We studied how the inhomogeneity of a sequence affects the phase transition that takes place at DNA melting. Unlike previous works, which considered thermodynamic quantities averaged over many different inhomogeneous sequences, we focused on precise sequences and investigated the succession of local openings that lead to their dissociation. For this purpose, we performed Transfer Integral type calculations with two different dynamical models, namely the heterogeneous Dauxois-Peyrard-Bishop model and the model based on finite stacking enthalpies we recently proposed. It appears that, for both models, the essential effect of heterogeneity is to let different portions of the investigated sequences open at slightly different temperatures. Besides this macroscopic effect, the local aperture of each portion indeed turns out to be very similar to that of a homogeneous sequence with the same length. Rounding of each local opening transition is therefore merely a size effect. For the Dauxois-Peyrard-Bishop model, sequences with a few thousands base pairs are still far from the thermodynamic limit, so that it is inappropriate, for this model, to discuss the order of the transition associated with each local opening. In contrast, sequences with several hundreds to a few thousands base pairs are pretty close to the thermodynamic limit for the model we proposed. The temperature interval where power laws holds is consequently broad enough to enable the estimation of critical exponents. On the basis of the few examples we investigated, it seems that, for our model, disorder does not necessarily induce a decrease of the order of the transition.

(2) email : Marc.JOYEUX@ujf-grenoble.fr

PACS numbers: 87.14.Gg, 05.70.Jk, 87.15.Aa, 64.70.-p
I. INTRODUCTION

This article is the last one of a series of three papers aimed at investigating the statistical physics of DNA denaturation, i.e. the separation of the two strands upon heating, on the basis of dynamical models like the Dauxois-Peyrard-Bishop one and models we recently proposed to take the finiteness of stacking interactions explicitly into account. In the first article of the series, we analyzed the denaturation of homogeneous sequences at the thermodynamic limit of infinitely long chains. We calculated the six fundamental exponents which characterize the critical behaviour of the specific heat, the order parameter, the correlation length, etc., by using the Transfer Integral (TI) technique. We showed that for the two investigated models the exponent for the specific heat is significantly larger than 1, which indicates that, within the validity of these models, denaturation is a first order phase transition. We also checked the validity of the four scaling laws which connect the six exponents and observed that Rushbrooke and Widom identities are satisfied, but not Josephson and Fisher ones. While the invalidation of Fisher identity is without any doubt a consequence of the dimensionality $d = 1$ of the investigated models, we argued that the failure of Josephson identity may well be due to the divergence of the order parameter, i.e. the average separation between paired bases.

The purpose of the second article of the series was to describe how the finite length of real sequences affects their critical properties. We characterized in some detail the three effects that are observed when the length of homogeneous sequences is decreased, namely, the decrease of the critical temperature, the decrease of the peak values of all quantities (like the specific heat and the correlation length) that diverge at the thermodynamic limit but remain finite for finite sequences, and the broadening of the temperature range over which the critical point affects the dynamics of the system. We furthermore performed a finite size scaling analysis of the models and showed that the singular part of the free energy can indeed be expressed in terms of a homogeneous function. We however pointed out that, because of the invalidation of Josephson identity, the derivation of the characteristic exponents which appear in the expression of the specific heat requires some care.

The investigations performed so far therefore dealt with homogeneous sequences. The reason is that homogeneous sequences display only one phase transition, that is, the whole sequence opens at a single well-characterized temperature on which theoretical inves-
tigations can focus. In contrast, the examination of UV absorption spectra revealed a long time ago that the denaturation of sufficiently long inhomogeneous sequences occurs through a series of local openings when temperature is increased \[16\], which makes this problem substantially more difficult to analyze. However, since all real DNA molecules display a heterogeneous, almost random-looking, distribution of A, T, G and C base pairs, the statistical physics description of the denaturation of such inhomogeneous sequences appears as a necessity.

In the language of statistical physics, a heterogeneous distribution of the individual components constituting a complex system is called *disorder*. One distinguishes *field disorder*, where heterogeneity concerns the distribution of the external field coupled to every component of the system, from *bond disorder*, which accounts for a heterogeneous distribution of the interactions between the elementary components of the system. No external field is considered in the present paper, which therefore focuses on bond disorder. The consequences of the introduction of disorder in a homogeneous system which displays a *second order* phase transition have been characterized by Harris \[17\]. According to Harris criterion, disorder does not affect the critical behaviour of the homogeneous system if the correlation length critical exponent \(\nu\) fulfills the inequality \(\nu \geq 2/d\), where \(d\) is the dimensionality of the system, because this implies that the correlation length is large enough to smear out heterogeneities close to the critical point. If Harris criterion is instead violated, then a new critical point generally sets in. The exponents of the power laws that are observed in the neighborhood of this new critical point satisfy Harris criterion. Harris work was extended a few years later by Imry and Wortis \[18\] to systems with a sharp *first order* phase transition at the homogeneous limit. On the basis of a heuristic argument, Imry and Wortis suggested that all first order transitions of homogeneous systems could well be rounded and transformed to second order transitions upon introduction of disorder, except if the dimensionality of the system is larger than a certain critical dimensionality \(d_c\) and its correlation length sufficiently large. Note, however, that Imry and Wortis’ argumentation explicitly assumes a finite correlation length at the critical temperature, while DNA melting corresponds to a somewhat peculiar first order phase transition with diverging correlation length. More recently, Hui and Berker \[19, 20\], and Aizenman and Wehr \[21\], showed on the basis of general arguments that if a temperature-driven first order phase transition involves a symmetry breaking, then it converts to a second order phase transition upon introduction of disorder. Otherwise, *i.e.* if the
critical point involves no symmetry breaking, then it is simply eliminated by disorder. Since DNA denaturation, as described by the models we investigate, does not involve symmetry breaking, this would imply that the denaturation of heterogeneous DNA sequences is neither associated with a phase transition nor a succession thereof.

Beside these general theoretical investigations, the question of the introduction of disorder in DNA sequences has been the subject of recent simulations [22, 23, 24, 25, 26], which dealt with models inspired from the Poland-Scheraga one [27] in the regime where the pure model displays a first-order transition, \( \text{i.e.} \) for a loop exponent \( c = 2.15 > 2 \). These studies lead to contradictory interpretations. Garel and Monthus [22, 23] indeed concluded that the transition remains first order in the disordered case, while Coluzzi and Yeramian [24, 25, 26] instead expressed the opinion that the random system undergoes a second order transition. It should be emphasized that these studies considered disorder-averaged thermodynamic observables and agreed on the point that these observables are not self-averaging at critical points, essentially because of the distribution of pseudo-critical temperatures over the ensemble of samples [22, 26]. In the present work, we will tackle a different question: is it sensible to describe the succession of local openings, which take place when the temperature of a precise heterogeneous sequence is increased, as a series of local phase transitions and, eventually, to specify the order of the local transitions?

The remainder of this paper is organized as follows. The dynamical models whose physical statistics we investigate are briefly described in Sec. II for the sake of completeness. We next derive in Sec. III the TI formulae which enable the calculation of the thermodynamic properties of finite heterogeneous sequences. We discuss in Sec. IV the critical behaviour of the specific heat per particle, \( c_V = C_V/N \), the average bubble depths, \( \langle y_n \rangle \), and the correlation length, \( \xi \), before concluding in Sec. V.

II. NONLINEAR HAMILTONIAN MODELS FOR INHOMOGENEOUS DNA SEQUENCES

The Hamiltonians of the two DNA models whose critical behavior is studied in this paper are of the form
\[ H = \sum_{n=1}^{N} \left\{ \frac{p_n^2}{2m} + V_M^{(n)}(y_n) + W^{(n)}(y_n, y_{n-1}) \right\}, \]  

(1)

where \( y_n \) is the transverse stretching at the \( n \)th pair of bases, \( V_M^{(n)}(y_n) \) describes the energy that binds the two bases of pair \( n \), and \( W^{(n)}(y_n, y_{n-1}) \) stands for the stacking interaction between base pairs \( n-1 \) and \( n \). The superscripts \( (n) \) in these terms indicate that both the on-site and the stacking interactions may be site-dependent for inhomogeneous sequences. The two models agree in representing the interbase bond \( V_M^{(n)}(y_n) \) by Morse potentials but the expressions for the stacking interactions are rather different. Moreover, the heterogeneous Dauxois-Peyrard-Bishop (DPB) model \[8, 14\] assumes that heterogeneity is essentially carried by different Morse parameters for AT and GC base pairs, while the models we proposed \[10, 11\] are based, like thermodynamic ones \[9\], on a set of ten different finite stacking enthalpies \( \Delta H_n \) corresponding to all possible oriented successions of base pairs. More precisely, for the heterogeneous DPB model \[8, 14\]

\[ V_M^{(n)}(y_n) = D_n \left( 1 - e^{-a_n y_n} \right)^2 \]

\[ W^{(n)}(y_n, y_{n-1}) = W(y_n, y_{n-1}) = \frac{K}{2} (y_n - y_{n-1})^2 \left[ 1 + \rho e^{-\alpha(y_n+y_{n-1})} \right], \]  

(2)

while for our model \[10\], hereafter called the JB model,

\[ V_M^{(n)}(y_n) = V_M(y_n) = D \left( 1 - e^{-a y_n} \right)^2 \]

\[ W^{(n)}(y_n, y_{n-1}) = \frac{\Delta H_n}{2} \left( 1 - e^{-b(y_n-y_{n-1})^2} \right) + K_b (y_n - y_{n-1})^2. \]  

(3)

The nonlinear stacking interaction in Eq. \(2\) has the particularity of having a coupling constant which drops from \( K(1 + \rho) \) to \( K \) as the paired bases separate. This decreases the rigidity of DNA sequences close to dissociation and results in a sharp first-order transition \[7\]. The first term in the expression of \( W^{(n)}(y_n, y_{n-1}) \) in Eq. \(3\) describes the finite stacking interaction and the second one the stiffness of the phosphate/sugar backbone. Introduction of finite stacking enthalpies \( \Delta H_n \) in the model is by itself sufficient to ensure a first-order denaturation transition \[10\].

We used two sets of numerical values for the DBP Hamiltonian. For the calculation of the melting profiles discussed in Sec. III, we used the set of parameters of Zhang et al. \[14\], that is, \( D_n = 0.038 \) eV for AT base pairs, \( D_n = 0.042 \) eV for GC base pairs, \( a_n = 4.2 \, \text{Å}^{-1} \) for both
AT and GC base pairs, $K = 0.042 \text{ eV} \cdot \text{Å}^{-2}$, $\rho = 0.5$, and $\alpha = 0.35 \text{ Å}^{-1}$. For the discussion of the specific heat critical exponent in Sec. IV, we instead used values that coincide, except for the $D_n$, with those we used in our work on critical exponents [12]. More explicitly, $D_n = 0.027 \text{ eV}$ for AT base pairs, $D_n = 0.033 \text{ eV}$ for GC base pairs, $a_n = 4.5 \text{ Å}^{-1}$ for both AT and GC base pairs, $K = 0.06 \text{ eV} \cdot \text{Å}^{-2}$, $\rho = 1.0$, and $\alpha = 0.35 \text{ Å}^{-1}$. The ten values of the stacking enthalpies $\Delta H_n$ of the JB model were taken from Table 1 of Ref. [9] and the other parameters of this model are those of Ref. [10], that is, $D = 0.04 \text{ eV}$, $a = 4.45 \text{ Å}^{-1}$, $K_b = 10^{-5} \text{ eV} \cdot \text{Å}^{-2}$ and $b = 0.10 \text{ Å}^{-2}$. Finally, the reduced mass of each base pair was considered to be $m = 300 \text{ uma}$ in the molecular dynamics simulations reported in the next section.

### III. TI CALCULATIONS FOR INHOMOGENEOUS DNA SEQUENCES

When ignoring the dissociation equilibrium $S_2 \leftrightarrow 2S$, which properly governs the separation of the two complementary strands ($S$) when the last base pair of double-stranded DNA ($S_2$) opens [4, 5, 8, 14], and neglecting the trivial term arising from kinetic energy, the partition function for the DNA models of Eq. (1) with open boundary conditions can be expressed as

$$Z = \int dy_1 dy_2 \cdots dy_N e^{-\beta \sum_n \left( V^{(n)}(y_n) + W^{(n)}(y_n, y_{n-1}) \right)} \cdot \cdot \cdot,$$

where $\beta = (k_B T)^{-1}$ is the inverse temperature. The TI method [13, 14] is a technique that allows for the efficient computation of $Z$. While this method was originally developed to investigate homogeneous sequences at the thermodynamic limit of infinitely long chains [13], Zhang et al. [14] have shown how it can be adapted to finite sequences described by the heterogeneous DPB model. It turns out that, because of the symmetric form of the Hamiltonian for the JB model, TI calculations are quite simpler for this model than for the DPB one. In this section, we first indicate the successive steps for calculating the partition function of the JB model and, consequently, its free energy, entropy, and specific heat. We next derive expressions for two-point correlation functions.

The first step for calculating $Z$ consists in rewriting Eq. (1) in the form...
\[ Z = \int dy_1 dy_2 \cdots dy_N e^{-\beta V_M(y_1)/2} K_2(y_2, y_1) K_3(y_3, y_2) \cdots K_N(y_N, y_{N-1}) e^{-\beta V_M(y_N)/2}. \] (5)

where the TI kernel \( K_n(y, x) \) for base pair \( n \) interacting with base pair \( n - 1 \) has the form

\[ K_n(y, x) = \exp \left[ -\beta \left\{ \frac{1}{2} V_M^{(n)}(y) + \frac{1}{2} V_M^{(n-1)}(x) + W^{(n)}(y, x) \right\} \right]. \] (6)

For the DPB model, \( K_n(y, x) \) is not symmetric (\( K_n(y, x) \neq K_n(x, y) \)) when base pairs \( n \) and \( n - 1 \) are different. Zhang et al. [14], who used the DBP model, therefore had to develop a symmetrization procedure that makes the all scheme more complex. In contrast, \( K_n(y, x) \) is symmetric (\( K_n(y, x) = K_n(x, y) \)) for the JB model, whatever the base pairs at positions \( n \) and \( n - 1 \), so that no additional symmetrization procedure is required. For the JB model, the essential difference between the procedures for homogeneous and inhomogeneous sequences consequently arises from the fact that ten different kernels need be considered, one for each possible succession of two base pairs [9]. The trick borrowed from method 2 of Zhang et al. [14] then consists in developing each kernel in a different orthonormal basis

\[ K_n(y, x) = \sum_i \lambda_i^{(n)} \Phi_i^{(n)}(y) \Phi_i^{(n)}(x), \] (7)

where the \( \{ \Phi_i^{(n)} \} \) and \( \{ \lambda_i^{(n)} \} \) are the eigenvalues and eigenvectors of the TI operator and satisfy the equation

\[ \int dx \ K_n(x, y) \Phi_i^{(n)}(x) = \lambda_i^{(n)} \Phi_i^{(n)}(y). \] (8)

By defining

\[ a_i^{(1)} = \int dy e^{-\beta V_M(y)/2} \Phi_i^{(2)}(y) \]
\[ a_i^{(N)} = \int dy e^{-\beta V_M(y)/2} \Phi_i^{(N)}(y) \]
\[ B_{ij} = \sqrt{\lambda_i^{(N)} \lambda_j^{(2)} a_i^{(N)} a_j^{(1)}} \]
\[ D_{ij}^{(n)} = \sqrt{\lambda_i^{(n-1)} \lambda_j^{(n)} \int dy \Phi_i^{(n-1)}(y) \Phi_j^{(n)}(y)} \] (9)

and substituting the kernel expansion of Eq. (7) into Eq. (5), the partition function can be rewritten in the form
\[ Z = \sum_{i_2, \ldots, i_N} B_{i_N i_2} D_{i_3 i_4}^{(3)} \ldots D_{i_{N-2} i_{N-1}}^{(N-1)} D_{i_{N-1} i_N}^{(N)}, \]  

or, equivalently,

\[ Z = \text{Tr} \left\{ BD^{(3)} D^{(4)} \ldots D^{(N-1)} D^{(N)} \right\}, \]

where \( B \) stands for the matrix with elements \( B_{ij} \), \( D^{(n)} \) for the matrix with elements \( D_{ij}^{(n)} \), and \( \text{Tr} \) indicates that one must take the trace of the product of matrices. Finally, the free energy \( F \), the entropy \( S \), and the specific heat \( C_V \) are obtained from \( Z \) according to

\[ F = -k_B T \ln(Z) \]
\[ S = -\frac{\partial F}{\partial T} \]
\[ C_V = -T \frac{\partial^2 F}{\partial T^2}. \]  

Calculation of intensive thermodynamical functions proceeds only similar lines. For example, the mean elongation of the \( n' \)th base pair can be written in the form

\[ \langle y_n \rangle = \frac{1}{Z} \int dy_1 dy_2 \ldots dy_N y_n e^{-\beta V_M(y_n)/2} K_2 (y_2, y_1) K_3 (y_3, y_2) \ldots K_N (y_N, y_{N-1}) e^{-\beta V_M(y_N)/2}. \]  

Defining

\[ b_i^{(1)} = \int dy e^{-\beta V_M(y)/2} \Phi_i^{(2)} (y) y \]
\[ b_i^{(N)} = \int dy e^{-\beta V_M(y)/2} \Phi_i^{(N)} (y) y \]
\[ C_{ij}^{(1)} = \sqrt{\lambda_i^{(N)} \lambda_j^{(2)}} a_i^{(N)} b_j^{(1)} \]
\[ C_{ij}^{(N)} = \sqrt{\lambda_i^{(N)} \lambda_j^{(2)}} b_i^{(1)} a_j^{(N)} \]
\[ Y_{1,ij}^{(n)} = \sqrt{\lambda_i^{(n-1)} \lambda_j^{(n)}} \int dy \Phi_i^{(n-1)} (y) \Phi_j^{(n)} (y) y, \]

and substituting Eq. (7) into Eq. (13), the mean elongation is obtained in the form

\[ \langle y_n \rangle = \frac{1}{Z} \text{Tr} \left\{ BD^{(3)} D^{(4)} \ldots D^{(n)} Y_{1}^{(n+1)} D^{(n+2)} \ldots D^{(N)} \right\}. \]
for \( n \neq 1 \) and \( n \neq N \), and
\[
\langle y_n \rangle = \frac{1}{Z} \text{Tr} \left\{ C^{(n)} D^{(3)} D^{(4)} \ldots D^{(N-1)} D^{(N)} \right\}
\] (16)
at the extremities of the chain, that is, for \( n = 1 \) or \( n = N \).

Two-point correlation functions are derived in the same manner. One obtains
\[
\langle y_n y_m \rangle = \frac{1}{Z} \text{Tr} \left\{ B D^{(3)} D^{(4)} \ldots D^{(m)} Y_1^{(m+1)} D^{(m+2)} \ldots D^{(n)} Y_1^{(n+1)} D^{(n+2)} \ldots D^{(N)} \right\}
\] (17)
if \( m \) and \( n \) are different from 1 and \( N \),
\[
\langle y_n^2 \rangle = \frac{1}{Z} \text{Tr} \left\{ E^{(nn)} D^{(3)} D^{(4)} \ldots D^{(N-1)} D^{(N)} \right\},
\] (18)
if \( m \) is different from 1 and \( N \) but \( n \) is equal to 1 or \( N \), and
\[
\langle y_1 y_N \rangle = \frac{1}{Z} \text{Tr} \left\{ E^{(1N)} D^{(3)} D^{(4)} \ldots D^{(N-1)} D^{(N)} \right\}.
\] (19)

In Eqs. (17)-(19) we noted
\[
\begin{align*}
c_i^{(1)} &= \int dy e^{-\beta V_M(y)/2} \Phi_i^{(2)}(y) y^2 \\
c_i^{(N)} &= \int dy e^{-\beta V_M(y)/2} \Phi_i^{(N)}(y) y^2 \\
E_{ij}^{(11)} &= \sqrt{\lambda_i^{(N)} \lambda_j^{(2)}} \frac{a_i^{(N)}}{a_j^{(1)}} \\
E_{ij}^{(1N)} &= \sqrt{\lambda_i^{(N)} \lambda_j^{(2)}} \frac{a_i^{(N)}}{a_j^{(1)}} \\
E_{ij}^{(NN)} &= \sqrt{\lambda_i^{(N)} \lambda_j^{(2)}} \frac{c_i^{(N)}}{a_j^{(1)}} \\
Y_{2,ij}^{(n)} &= \sqrt{\lambda_i^{(n-1)} \lambda_j^{(n)}} \int dy \Phi_i^{(n-1)}(y) \Phi_j^{(n)}(y) y^2.
\end{align*}
\] (20)

In order to check the accuracy of the TI procedure, we compared melting profiles obtained with this method to those obtained from molecular dynamics (MD) simulations. MD simulations consist in integrating numerically Langevin equations of motion.
\[
\frac{m}{dt^2} y_n = -\frac{\partial H}{\partial y_n} - m\gamma \frac{dy_n}{dt} + \sqrt{2mk_B T} w(t).
\]  
(21)

The second and third term in the right-hand side of this equation model the effects of the solvent on the sequence. \( \gamma \) is the dissipation coefficient (we assumed \( \gamma = 5 \text{ ns}^{-1} \) as in Refs. \[10, 11, 28\]) and \( w(t) \) a normally distributed random function with zero mean value and unit variance. Step by step integration, with 10 fs steps, was performed by applying a second order Brünger-Brooks-Karplus integrator \[29\] to the sequence initially at equilibrium at 0 K and subjected to a temperature ramp of 10 K/ns. This slow heating insures that the temperature of the system, estimated from its average kinetic energy

\[
T_{\text{kin}} = \frac{2}{Nk_B} \sum_{n=1}^{N} \frac{p_n^2}{2m},
\]  
(22)

closely follows the temperature \( T \) imposed by the random kicks. Once the required temperature was reached, Langevin equations were integrated at constant temperature for additional 30 ns in order to bring the system still closer to thermal equilibrium. We finally averaged the base pair separations \( y_n \) over time intervals which varied between 1 \( \mu \)s for temperatures substantially smaller than the melting one, up to 5 \( \mu \)s close to melting, in order to correctly average the low frequency thermal fluctuations which develop close to the critical point \[28\]. During the averaging process, we went on recording the physical temperature of the system (Eq. (22)), because its final agreement with the imposed temperature \( T \) provides an estimate of the quality of the averaging. For all the results presented below, the differences between the two temperatures were kept below 0.1 K.

Figs. 1 and 2 show the melting profiles \( \langle y_n \rangle \) as a function of \( n \) at increasing temperatures for, respectively, the 1793 base pairs (bp) human \( \beta \)-actin cDNA (NCB entry code NM_00110) and the 2399 bp inhibitor of the hepatocyte growth factor activator \[30\], which were obtained from TI calculations with the JB model. In contrast with the estimation of critical exponents \[12, 15\], this kind of plots does not require a very high precision, so that the grid on which the matrix representations of the TI kernels \( K_n(y, x) \) were built \[13\] consisted of only 2901 \( y \) values regularly spaced between \( y_{\text{min}} = -100/a \) and \( y_{\text{max}} = 2800/a \) with steps of 1/a. The melting profiles for the actin sequence at 322 K and 346 K obtained from TI calculations and MD simulations performed with the JB model are compared in Fig. 3. It is seen that even tiny details coincide for the two curves at 322 K (bottom plot). The agreement
remains excellent closer to denaturation. In particular, both methods conclude that all base pairs with \( n > 1200 \) are open at this temperature. We will come back to this point later. Note that resolution with respect to base pair positions is, however, substantially higher in TI results, although TI calculations were more rapid than MD simulations by a factor of almost 10 close to melting. In spite of the fact that the TI procedure is much more CPU demanding for inhomogeneous sequences than for homogeneous ones, it therefore still appears as a very powerful tool compared to MD simulations. Fig. 4 compares melting profiles for the actin sequence at 350 K obtained from TI and MD calculations performed with the heterogeneous DPB model. Although the agreement is again excellent, it is seen that the TI profile looks like as if it consisted of 3 or 4 superposed curves. This is most probably due to the conjunction of two phenomena: (i) the heterogeneous DBP model assumes that the Morse interaction for GC base pairs is stronger than that for AT base pairs, and (ii) the resolution of the TI procedure is high enough to reflect the variations of \( \langle y_n \rangle \) at the level of single base pairs that result from this difference. To confirm this hypothesis, we checked that the same phenomenon does show up for the JB model. Still, since this model assumes that heterogeneity is carried by stacking interactions instead of on-site potentials, superposed curves essentially appear in the plots of \( \langle y_n - y_{n-1} \rangle \) as a function of \( n \). Moreover, the phenomenon is somewhat attenuated compared to Fig. 4, because the JB model considers ten different stacking enthalpies, while the DPB one considers only two different Morse potential strengths. Finally, Fig. 5 shows the melting curve, that is, the evolution with temperature of the portion of open base pairs, for the 1793 bp actin sequence obtained with the JB model. Although they were computed with different models, this curve compares very well with the one drawn in Fig. 4 of Ref. \[11\].

In conclusion, the TI procedure appears as a powerful and trustful tool for the computation of the thermodynamic properties of inhomogeneous DNA sequences.

**IV. EFFECTS OF DISORDER CLOSE TO MELTING**

In this section, we will investigate the role of disorder close to the critical point. In contrast with previous studies \[22, 23, 24, 25, 26\], we will not consider disorder-averaged quantities, that is, we will not discuss the statistical physics of an ensemble of random sequences. Instead, we will focus on precise sequences and try to determine if the successive
openings that lead to the dissociation of these sequences may be described as phase transitions, and eventually address the question of the order of these transitions. To this end, we will study the behaviour of the specific heat per particle, \( c_V = C_V/N \), the average bubble depths, \( \langle y_n \rangle \), and the correlation length, \( \xi \), close to the critical temperature.

A. Critical behaviour of \( c_V \)

The evolution of \( c_V \) with temperature was computed for the 1793 bp actin and the 2399 bp inhibitor according to Eqs. (11) and (12). Finite differences were used to estimate the second derivative of \( Z \) in Eq. (12). The results obtained with grids of 2901 values of \( y \) regularly spaced between \(-100/a\) and \(2800/a\) are shown in Fig. 6. The evolution of \( c_V \) in these plots is most easily understood when comparing them to the corresponding profiles in Figs. 1 and 2. The bottom plot in Fig. 1 indeed indicates that the average AT content for the 1793 bp actin sequence is substantially higher for base pairs with \( n > 1150 \). It is seen in the top plot of Fig. 1, that one third of the sequence (the base pairs with \( n > 1150 \)) consequently melts around 346-348 K, while the remaining two thirds (the base pairs with \( n \leq 1150 \)) melt at the slightly higher temperature of about 354 K. This two-steps denaturation is perfectly reflected in the temperature evolution of \( c_V \) (top plot of Fig. 6), which displays two peaks with 1:2 relative intensities centred around 348 and 354 K. For the 2399 bp inhibitor, the bottom plot of Fig. 2 similarly indicates that the average AT content is rather uniform in the sequence, except that it significantly decreases with decreasing \( n \) for the first 600 base pairs. Not surprisingly, it is accordingly seen in the top plot of Fig. 2 that these first 600 base pairs melt about three degrees above the temperature of 352-354 K where the rest of the sequence dissociates. Since this second melting step involves only about one fourth of the sequence and takes place very close to the first step, it merely appears as a shoulder on the high temperature side of the plot of \( c_V \) in the bottom plot of Fig. 6.

In order to learn more about these openings, we next draw log-log plots of the evolution of \( c_V \) as a function of the reduced temperature \( t \), defined according to

\[
t = 1 - \frac{T}{T_c}.
\]

In the case of homogeneous sequences, the critical temperature \( T_c \) that appears in Eq. (23) is unambiguously defined. This is no longer the case when dealing with inhomogeneous
sequences, so that in the following we will explicitly state which temperature is used as $T_c$. Moreover, this kind of plot requires more precision than the previous figures. The calculation of $Z$ in Eq. (11) was therefore performed with grids of 4101 values of $y$ regularly spaced between $-100/a$ and $4000/a$. The result obtained for the JB model and the 2399 bp inhibitor is displayed in the bottom plot of Fig. 7. The solid line shows the result for the 2399 bp inhibitor sequence, while the dashed and dot-dashed lines show results that we previously obtained for a 2000 bp homogeneous sequence and an infinitely long homogeneous sequence, respectively (see the bottom plot of Fig. 3 of Ref. [15]). For the inhomogeneous sequence, $T_c$ was taken as the temperature where $c_V$ is maximum (for the grid with 4201 points, we numerically obtained $T_c = 354.34$ K), so that the solid line actually deals with the first step of the melting of the inhibitor sequence, that is, the opening of the base pairs with $n > 600$. In Ref. [15], we arrived to the conclusion that the thermodynamics of sequences with a few thousands base pairs are close to that of infinite ones down to $t \approx 10^{-3}$ for the JB model. As a consequence, the curves for the 2000 bp and infinitely long homogeneous sequences are almost superposed above this threshold. Stated in other words, the rounding of the phase transition is hardly noticeable for temperatures which differ from the critical one by more than a few tenths of a degree. Examination of the bottom plot of Fig. 7 further shows that the thermodynamics of the opening of the 1800 base pairs with $n > 600$ of the inhibitor sequence is also very similar to that of the finite ($N = 2000$) and infinite homogeneous sequences: rounding is indeed imperceptible about one degree ($t \approx 3.10^{-3}$) below the critical temperature. The power law dependence of $c_V$ against $t$ therefore extends over an interval of $t$ values which is sufficiently large to allow for the estimation of the critical exponent $\alpha$ of $c_V$. One obtains $\alpha = 1.07$, which is characteristic of a first order phase transition.

The top plot of Fig. 7 also displays a log-log plot of the evolution of $c_V$ with $t$ computed, however, with the heterogeneous DBP model. For the grid with 4201 points and this model, we found $T_c = 284.24$ K. We showed in Ref. [15] that, in contrast with the JB model, sequences with $N = 2000$ bp are still far from the thermodynamic limit for the DBP model. Therefore, the dashed curve (homogeneous sequence with $N = 2000$ bp) and the dot-dashed one (homogeneous sequence at the thermodynamic limit) are well separated. Examination of this plot also indicates that the (solid) curve for the inhomogeneous 2399 bp inhibitor sequence is again qualitatively close to the (dashed) curve for the homogeneous
2000 bp sequence - and consequently quite separated from the curve for the sequence at the thermodynamic limit.

One might therefore tentatively conclude from the results presented in this subsection that, for a given sequence, the essential effect of heterogeneity is to let different portions of the sequence open at slightly different temperatures. Besides this global effect, the dynamics of the local aperture of each portion is indeed very similar to that of a homogeneous sequence with the same length. We will now investigate the critical behaviour of the depth of the bubbles and of the correlation length, in order to check whether they confirm this conclusion.

B. Critical behaviour of the bubble depth \( \langle y_n \rangle \)

As we already noted, the 1793 bp actin sequence opens in two fairly separated steps: the base pairs with \( n > 1150 \) melt around 348 K, while those with \( n < 1150 \) melt at the slightly higher temperature of 354 K (see Figs. 1, 5 and 6). Finer details can be observed in Fig. 1. It is indeed seen that melting of the \( n > 1150 \) portion is driven by three bubbles centred around \( n = 1300 \), \( n = 1450 \) and \( n = 1610 \), while melting of the \( n < 1150 \) portion is driven by two bubbles centred around \( n = 318 \) and \( n = 441 \), the centre of each bubble corresponding to a local maximum of the AT percentage. Fig. 8 displays log-log plots of the average depth of each bubble, \( \langle y_n \rangle \), as a function of the reduced temperature \( t \), obtained with the JB model. For the three bubbles with \( n > 1150 \) (top plot), the critical temperature was taken as the temperature \( T_c = 348.2 \) K of the secondary maximum of the specific heat, while for the two bubbles with \( n < 1150 \) (bottom plot), the critical temperature was taken as the temperature \( T_c = 353.9 \) K of the principal maximum of \( c_V \). Fig. 8 indicates that (i) the average depth of all bubbles exhibits a power law dependence against \( t \) over a reasonably large interval of temperatures, (ii) the slopes are essentially identical for all bubbles belonging to the same portion of the sequence, and (iii) the critical exponents that can be deduced from these slopes, that is, -1.28 for the bubbles with \( n > 1150 \) and -1.00 for the bubbles with \( n < 1150 \), are close to the critical exponent \( \beta = -1.31 \) we obtained at the thermodynamic limit [12].
C. Critical behaviour of the correlation length $\xi$

At the thermodynamic limit of infinitely long chains, the two-point spatial autocorrelation function

$$C_{ij} = \langle y_i y_j \rangle - \langle y_i \rangle \langle y_j \rangle$$  \hspace{1cm} (24)

varies for large values of $|i - j|$ according to

$$C_{ij} \propto \exp\left(-\frac{|i - j|}{\xi}\right),$$  \hspace{1cm} (25)

where $\xi$ is the correlation length. $\xi$ can consequently be obtained as the inverse of the slope in the plots of $\ln(C_{ij})$ as a function of $|i - j|$. Such plots are shown in Fig. 9 for a homogeneous sequence with 10000 base pairs described with the homogeneous version of the JB model [10, 12, 15]. It is seen that the natural logarithm of $C_{ij}$ indeed evolves linearly with $j - i$ over more than 20 orders of magnitudes and that the correlation length $\xi$ can be determined very accurately from the slope of these curves. When plotting the values of $\xi$ obtained in this way as a function of $t$ (critical temperature is $T_c = 367.47$ K), one furthermore recovers the critical exponent $\nu = 1.23$ reported in Ref. [12]. Similar plots of $\ln(C_{ij})$ as a function of $j - i$, obtained from Eqs. (15), (17), and (24), are reported in Fig. 10 for the 1793 bp actin sequence described with the JB model. The main plot was obtained by setting $i = 180$ and the smaller one by setting $i = 1250$. The horizontal and vertical scales are identical for both plots, but the smaller one ($i = 1250$) was horizontally shifted so that identical values of $j$ are vertically aligned. Examination of Fig. 10 indicates that all curves in the main plot and some curves in the smaller plot are composed of two segments instead of a single straight line, and that the values of $j$ where the two segments cross approximately coincide, for each temperature, with the boundary between the double-stranded and open portions of the sequence. Moreover, local slopes are much smaller whenever $i$ and/or $j$ lie in the open portion of the sequence. By comparing the two plots in Fig. 10, one finally notices that absolute values of $\ln(C_{ij})$ are different for different values of $i$, but that their variations are identical for identical values of $j$. These two observations suggest that for inhomogeneous sequences the two-point spatial correlation function $C_{ij}$ still evolves exponentially with $|i - j|$, as in Eq. (25), but that there exists one different correlation length $\xi$ for each region that melts independently from the rest of the sequence. Note that it is then quite appropriate
to call these regions *coherence regions*. At last, we checked that the correlation lengths obtained from the slopes of the first segments in the main plot of Fig. 10 evolve as $t^{-1.13}$ ($T_c = 353.9$ K, as in the bottom plot of Fig. 8). Therefore, the correlation length critical exponent for the portion of the sequence with $n < 1150$ is again close to the above mentioned value $\nu = 1.23$ for homogeneous sequences [12].

V. CONCLUSION

In this work, we analyzed the statistical physics of inhomogeneous DNA sequences close to denaturation. Unlike previous studies, which considered disorder-averaged thermodynamic observables, we focused on the successive local openings of precise sequences. To this end, we used the extended TI method of Zhang et al [14] to investigate the properties of the heterogeneous DPB model [8], and derived a modified version of this method to adapt it to the study of the JB model [10, 12, 15]. Examination of the critical behaviour of the specific heat per particle, $c_V$, the average bubble depths, $\langle y_n \rangle$, and the correlation length, $\xi$, leads to the following conclusions. Both models agree in pointing out that the principal effect of heterogeneity is to let different portions of the sequence open at slightly different temperatures. Besides this global effect, the dynamics of the local aperture of each portion is indeed very similar to that of a homogeneous sequence with the same length. In particular, the local melting transition of each portion is rounded by finite size effects [15]. Strictly speaking, one should therefore not describe the melting of an inhomogeneous sequence as a succession of phase transitions. When speaking more loosely, such a description is however not really wrong, in the sense that the melting of several hundreds or a few thousands of base pairs is accompanied by a sharp maximum of the specific heat and a clear step of the entropy (see Fig. 6 and Figs. 2 and 3 of Ref. [15]). The answer to the more involved question concerning the possibility to ascribe an order to these rounded transitions unfortunately turns out to depend on the model which is used to describe DNA. Indeed, for the JB model, sequences (or portions thereof) with several hundreds to a few thousands base pairs are already rather close to the thermodynamic limit (see the bottom plot of Fig. 7 and Figs. 3 and 4 of Ref. [15]), so that power laws are observed over significant temperature intervals. For the 2399 bp inhibitor and the 1793 bp actin sequences, the values of the critical exponents estimated on these temperature intervals turn out to be close to those of
homogeneous sequences at the thermodynamic limit. In particular the specific heat critical exponent we obtained for the opening of the 1800 base pairs with \( n > 600 \) of the inhibitor sequence, \( \alpha = 1.07 \), is characteristic of a first order phase transition. Of course, it is not possible to draw a general conclusion from a single example, but this calculation still has the merit of showing that disorder does not necessarily reduce the order of the transition. In contrast, for the DPB model, sequences with a few thousands base pairs are still quite far from the thermodynamic limit (see the top plot of Fig. 7 and Fig. 3 of Ref. [15]), so that it is not appropriate to discuss the order of the melting transition for inhomogeneous sequences described by this model.

Last but not least, it should be emphasized that the two Morse parameters \( D_n \) for AT and GC pairing and the ten stacking enthalpies \( \Delta H_n \) cannot be extracted independently from experimental denaturation curves [31, 32, 33]. It has however been shown recently how these twelve quantities can be obtained from the properties of nicked DNA [32, 33]. The free energies reported in Table 1 of Ref. [33] indicate that heterogeneity in improved dynamical models of DNA secondary structure should be carried by both pairing and stacking energies. It will therefore be very instructive to build a dynamical model centred on these data and check whether the description of the melting phase transition of inhomogeneous DNA obtained from this model matches that obtained with the DPB or the JB models (note that the new parameters have already been used in statistical models, see [34]). Aside from the adjustment of the remaining free parameters of the model against experimental melting curves, the major difficulty of this task will consist in establishing a TI calculation procedure that allows to take into account the heterogeneity of both pairing and stacking energies.

[1] R. Thomas, Bull. Sté. Chim. Biol. 35, 609 (1953)
[2] R. Thomas, Biochim. Biophys. Acta 14, 231 (1954)
[3] J. Marmur, R. Rownd and C.L. Schildkraut, Prog. Nucleic Acid Res. 1, 231 (1963)
[4] D. Poland and H.A. Scheraga, Theory of Helix-Coil Transitions in Biopolymers (Academic Press, New York, 1970)
[5] R.M. Wartell and A.S. Benight, Phys. Rep. 126, 67 (1985)
[6] M. Peyrard, Nonlinearity 17, R1 (2004)
[7] T. Dauxois, M. Peyrard and A.R. Bishop, Phys. Rev. E 47, R44 (1993)
[8] A. Campa and A. Giansanti, Phys. Rev. E 58, 3585 (1998)
[9] R.D. Blake, J.W. Bizzaro, J.D. Blake, G.R. Day, S.G. Delcourt, J. Knowles, K.A. Marx and J. SantaLucia, Bioinformatics, 15, 370 (1999)
[10] M. Joyeux and S. Buyukdagli, Phys. Rev. E 72, 051902 (2005)
[11] S. Buyukdagli, M. Sanrey and M. Joyeux, Chem. Phys. Letters 419, 434 (2006)
[12] S. Buyukdagli and M. Joyeux, Phys. Rev. E 73, 051910 (2006)
[13] T. Schneider and E. Stoll, Phys. Rev. B 22, 5317 (1980)
[14] Y.-L. Zhang, W.-M. Zheng, J.-X. Liu and Y.Z. Chen, Phys. Rev. E 56, 7100 (1997)
[15] S. Buyukdagli and M. Joyeux, Phys. Rev. E 76, 021917 (2007)
[16] O. Gotoh, Adv. Biophys. 16, iii (1983)
[17] A.B. Harris, J. Phys. C 7, 1671 (1974)
[18] Y. Imry and M. Wortis, Phys. Rev. B 19, 3580 (1979)
[19] K. Hui and A.N. Berker, Phys. Rev. Lett. 62, 2507 (1989); 63, 2433 (E) (1989)
[20] A.N. Berker, J. Appl. Phys. 70, 5941 (1991)
[21] M. Aizenman and J. Wehr, Phys. Rev. Lett. 62, 2503 (1989)
[22] C. Monthus and T. Garel, Eur. Phys. J. B 48, 393 (2005)
[23] T. Garel and C. Monthus, J. Stat. Phys. : Theory and Experiment, P06004 (2005)
[24] B. Coluzzi, Phys. Rev. E 73, 011911 (2006)
[25] B. Coluzzi and E. Yeramian, Philosophical Magazine 87, 517 (2007)
[26] B. Coluzzi and E. Yeramian, Eur. Phys. J. B 56, 349 (2007)
[27] D. Poland and H.A. Scheraga, J. Chem. Phys. 45, 1464 (1966)
[28] M. Joyeux, S. Buyukdagli and M. Sanrey, Phys. Rev. E 75, 061914 (2007)
[29] A. Brünger, C.B. Brooks and M. Karplus, Chem. Phys. Lett. 105, 495 (1984)
[30] T. Shimomura, K. Denda, A. Kitamura, T. Kawaguchi, M. Kito, J. Kondo, S. Kagaya, L. Qin, H. Takata, K. Miyazawa and N. Kitamura, J. Biol. Chem. 272, 6370 (1997)
[31] J. SantaLucia, Proc. Natl. Acad. Sci. USA 95, 1460 (1998)
[32] E. Protozanova, P. Yakovchuk and M.D. Frank-Kamenetskii, J. Mol. Biol. 342, 775 (2004)
[33] A. Krueger, E. Protozanova and M.D. Frank-Kamenetskii, Biophys. J. 90, 3091 (2006)
[34] T. Ambjornsson, S.K. Banik, O. Krichevsky and R. Metzler, Phys. Rev. Lett. 97, 128105 (2006)
FIGURE CAPTIONS

Figure 1: (color online) : (Top) plot, for increasing temperatures, of $\langle y_n \rangle$ as a function of the site number $n$ for the 1793 bp human $\beta$-actin cDNA sequence (NCB entry code NM_001101). These curves were obtained from TI calculations performed with the JB model. (Bottom) plot, as a function of $n$, of the AT percentage averaged over 40 consecutive bp of the actin sequence.

Figure 2: (color online) (Top) plot, for increasing temperatures, of $\langle y_n \rangle$ as a function of the site number $n$ for the 2399 bp inhibitor of the hepatocyte growth factor activator sequence [30]. These curves were obtained from TI calculations performed with the JB model. (Bottom) plot, as a function of $n$, of the AT percentage averaged over 40 consecutive bp of the inhibitor sequence.

Figure 3: (color online) Comparison of $\langle y_n \rangle$ profiles for the 1793 bp actin sequence at 322 K (bottom plot) and 346 K (top plot) obtained from TI calculations (dashed lines) and MD simulations (solid lines) performed with the JB model. The main plots show the profile of the whole sequence, while the inserts zoom in on 300 base pairs.

Figure 4: (color online) Comparison of $\langle y_n \rangle$ profiles for the 1793 bp actin sequence at 350 K obtained from TI calculations (small crosses) and MD simulations (solid line) performed with the heterogeneous DPB model. The main plot shows the profile of the whole sequence, while the insert zooms in on 300 base pairs.

Figure 5: (color online) Plot of the fraction of open base pairs as a function of temperature $T$ for the 1793 bp actin sequence, obtained from TI calculations performed with the JB model. The criterion for a base pair $n$ to be open is that $\langle y_n \rangle$ be larger than the threshold of 10 Å.

Figure 6: (color online) Plots of the specific heat per particle $c_V$ as a function of temperature $T$ for the 1793 bp actin sequence (top plot) and the 2399 bp inhibitor sequence (bottom plot), obtained from TI calculations performed with the JB model. $c_V$ is expressed in units of the Boltzmann constant $k_B$. 
**Figure 7**: (color online) Log-Log plots of the specific heat per particle $c_V$ as a function of the reduced temperature $t$ for the 2399 bp inhibitor sequence (solid lines), a 2000 bp homogeneous sequence (dashed lines), and an infinitely long homogeneous sequence (dot-dashed lines), obtained from TI calculations performed with the JB model (bottom plot) and the DPB model (top plot). $c_V$ is expressed in units of the Boltzmann constant $k_B$.

**Figure 8**: (color online) Log-Log plots, as a function of the reduced temperature $t$, of the average depth $\langle y_n \rangle$ of bubbles centred around $n = 1300$, $n = 1450$ and $n = 1640$ (top plot), and $n = 318$ and $n = 441$ (bottom plot) for the 1793 bp actin sequence. These results were obtained from TI calculations performed with the JB model. The critical temperature of each portion of the sequence is indicated on the corresponding plot.

**Figure 9**: (color online) Plots of $\ln(C_{ij})$ as a function of $|i - j|$ for a homogeneous sequence with 10000 base pairs at several temperatures comprised between 340 K and 367.2 K. $i = 1$ for all the plots. These results were obtained from TI calculations performed with the homogeneous JB model. Critical temperature of this system is $T_c = 367.47$ K.

**Figure 10**: (color online) Plots of $\ln(C_{ij})$ as a function of $|i - j|$ for the 1793 bp actin sequence at several temperatures regularly spaced between 340 K and 350 K. $i = 180$ for the main plot and $i = 1250$ for the smaller vignette. The horizontal and vertical scales are identical for both plots, but the vignette ($i = 1250$) was horizontally shifted so that identical values of $j$ are vertically aligned. These results were obtained from TI calculations performed with the JB model.
$<y_n>$ mean value (Å) AT percentage base position

- $<y_n>$ vs. base position $n$
- AT percentage distribution
The graph shows the mean value \( <y_n> \) (Å) as a function of base position \( n \). The graph is divided into two sections:

**346 K**
- The graph represents the mean value \( <y_n> \) at 346 K.
- There are two lines: one for the TI method (dashed red) and one for MD simulation (solid blue).

**322 K**
- The graph represents the mean value \( <y_n> \) at 322 K.
- Again, there are two lines: one for the TI method (dashed red) and one for MD simulation (solid blue).

The x-axis represents the base position \( n \), while the y-axis represents the mean value \( <y_n> \) (Å).
1793 bp actin
TI method
10 Å threshold
specific heat \( c_V \) (units of \( k_B \)) vs. temperature \( T \) (K)

- **1793 bp actin**
- **2399 bp inhibitor**
2399 bp inhibitor
2000 bp homogeneous
thermodynamic limit

specific heat $c_V$ (units of $k_B$)

DPB model

JB model

t $^{-1.07}$

reduced temperature $t$

$V$ (units of $k_B$)
Reduced temperature $t$ is $<y_n>$ mean value (Å) versus $t^{-1.28}$ for $T_c = 348.2$ K.

Reduced temperature $t$ is $<y_n>$ mean value (Å) versus $t^{-1.00}$ for $T_c = 353.9$ K.

$n=1300$, $n=1450$, $n=1610$, $n=318$, $n=441$. 

Citation: Reduced temperature $t$ is $<y_n>$ mean value (Å) versus $t^{-1.28}$ for $T_c = 348.2$ K.

Citation: Reduced temperature $t$ is $<y_n>$ mean value (Å) versus $t^{-1.00}$ for $T_c = 353.9$ K.
\[
\ln(C_{ij}) = i - T_c = 367.47 \text{ K}
\]
\[ \ln(C_{ij}) \]

For the 1793 bp actin, with:

- \( i = 180 \)
- \( i = 1250 \)