Classical double-layer atoms: artificial molecules

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The groundstate configuration and the eigenmodes of two parallel two-dimensional classical atoms are obtained as function of the interatomic distance \((d)\). The classical particles are confined by identical harmonic wells and repel each other through a Coulomb potential. As function of \(d\) we find several structural transitions which are of first or second order. For first (second) order transitions the first (second) derivative of the energy with respect to \(d\) is discontinuous, the radial position of the particles changes discontinuously (continuously) and the frequency of the eigenmodes exhibit a jump (one mode becomes soft, i.e. its frequency becomes zero).

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In recent years, there has been an increased interest in the study of finite two-dimensional (2D) systems of charged particles \([1]\). These quantum dots are atomic like structures which have interesting optical properties and may be of interest for single electron devices. Most of the previous works have treated the quantum mechanical problem of a small number of electrons. If the number of electrons is increased beyond \(6-7\) the Coulomb correlations, we have made a throughout study of the classical system \([2]\) in which the particles are taken point-like but where no approximation was made on the Coulomb correlation. This approach is valid in high magnetic fields where the kinetic energy of the electrons is quenched. In the ground state the charged particles are located on rings and a table of Mendeljev was constructed in Ref. \([2]\). Such classical confined systems have been observed in electron dimples on liquid helium \([3]\), in drops of colloidal suspensions \([4]\) and in confined dusty particles \([5]\).

Here we extend our previous work to the case of classical artificial molecules \([6]\) which consist of two classical 2D atoms which are laterally separated by a distance \(d\). Intuitively, we expect interesting behavior as function of \(d\) which governs the inter atomic interaction. This can be seen as follows, for \(d = 0\) we have just one 2D atom (e.g. for \(2N = 10\), this is a two ring structure), while for \(d \to \infty\) the system consists of two independent 2D atoms with each half of the total number of particles (e.g. two atoms each having 5 particles on one ring). This implies that as function of \(d\) structural transitions (e.g. configurational changes) have to take place. We found that these artificial molecules show a surprisingly complex behavior as function of \(d\). In this letter, we investigate the groundstate energy and its derivatives, the groundstate configurations and the normal modes, i.e. the eigenfrequencies, of the system.

We limit ourselves to the system consisting of an even number, \(2N\), of charged particles which are evenly distributed over two layers separated a distance \(d\). In both layers, the confinement potential is centered around the \(z\)-axis and this parabolic confinement, which is taken the same for both layers, keeps the system together in the \(xy\)-plane. We focus our attention on systems described by the Hamiltonian (where \(m\) is the mass of the particles, \(\omega_0\) the radial confinement frequency, \(e\) the particle charge, and \(\epsilon\) the dielectric constant of the medium the particles are moving in)

\[
H = H_I + H_{II} + H_{I,II},
\]

with

\[
H_S = \sum_{i \in \delta} \frac{1}{2} m \omega_0^2 r_i^2 + \frac{e^2}{\epsilon} \sum_{i < j \in \delta} \frac{1}{|\vec{r}_i - \vec{r}_j|},
\]

the confinement and interaction energy of the artificial atom in layer \(S = I, II\), and

\[
H_{I,II} = \frac{e^2}{\epsilon} \sum_{i \in I, j \in II} \frac{1}{|\vec{r}_i - \vec{r}_j|}.
\]

the interaction energy between the atoms in the two layers. For convenience, we will refer to our charged particles as electrons, keeping in mind that they can also be ions with charge \(e\) and mass \(m\). The Hamiltonian can be written in a dimensionless form if we express the coordinates, energy, and frequency in the following units \(r' = (e^2/\epsilon)^{1/3} \alpha^{-1/3}, E' = (e^2/\epsilon)^{2/3} \alpha^{1/3}, \omega' = \omega_0/\sqrt{2}\), respectively, with \(\alpha = m \omega_0^2 / 2\). The numerical values for the parameters \(r'\) and \(E'\) for some typical experimental systems were given in Ref. \([6]\). In dimensionless units the Hamiltonian becomes

\[
H = \sum_{i \in I} r_i^2 + \sum_{i < j \in I} \frac{1}{|\vec{r}_i - \vec{r}_j|} + \sum_{i \in II} r_i^2
\]

\[
+ \sum_{i < j \in II} \frac{1}{|\vec{r}_i - \vec{r}_j|} + \sum_{i \in I, j \in II} \frac{1}{|\vec{r}_i - \vec{r}_j|}.
\]
and consequently the groundstate energy is only a function of the number of electrons, $2N$, and the distance $d$ between the layers.

The numerical method used in the present study to obtain the groundstate configuration is based on the Monte Carlo technique supplemented with the Newton Method in order to increase the accuracy of the energy of the groundstate configuration. The latter technique is outlined and compared with the Monte Carlo technique in Ref. [1] and also yields the eigenfrequencies and the eigenmodes of the groundstate configuration.

Let us consider first the case of a molecule consisting of six electrons distributed over two atoms each with three electrons. For $d = 0$ this is a 2D atom with 6 electrons, of which we know the groundstate configuration [2], namely (1,5): 5 electrons on a ring and one electron in the center of the ring. In the opposite limit, $d \to \infty$, we have two independent 2D atoms, each consisting of three electrons for which the groundstate configuration consists of one ring containing three electrons [3]. Thus, as function of $d$, we expect a structural transition.

Fig. 1 shows the energy per electron of the groundstate and its first derivative with respect to $d$. At $d = 0.35955$ the first derivative is discontinuous and a structural transition takes place. For $d < 0.35955$ the groundstate configuration is (1,2)/(0,3): the configuration (1,2) in one layer (indicated by the open dots in the insets of Fig. 1) and (0,3) in the other layer (indicated by the solid dots). As viewed from above we have (1,5) which is the configuration of one atom consisting of six electrons. This implies that the inter-layer correlations are sufficiently strong to impose the one atom configuration to the $2N = 6$ electrons in the molecular structure. For $d > 0.35955$ the configuration is twice (0,3), which is the configuration of two independent atoms each consisting of three electrons.

The frequencies of the normal modes of the groundstate configuration are shown in the inset of Fig. 1. Notice that at the first order transition point the frequencies exhibit a jump. For $d > 0.35955$ more modes are degenerate in energy and in the limit of $d \to \infty$ all eigenfrequencies are at least twofold degenerate. The latter is a consequence of the fact that the electrons in one layer can vibrate in phase and out-of-phase with respect to the electrons in the other layer. With decreasing $d$ the inter-layer interaction destroys this degeneracy.

In Fig. 1 also the energies of the metastable states are shown. For $d = 0$ there is just one metastable state with configuration (0,6). If $d$ is slightly different from zero, there are four metastable states plus the groundstate. One obtains these configurations by moving one electron from one layer to the other without changing its lateral position, and this for each of the two configurations [(1,5) and (0,6)]. From Fig. 1 one sees clearly how a configuration, metastable for $d < 0.35955$, becomes the stable one for $d > 0.35955$.

The artificial molecule consisting of two times five electrons is far more complex and exhibits several transitions some of which are qualitatively different. For $d = 0$ this is a 2D atom with 10 electrons and the groundstate configuration is (2,8) [3]. For $d \to \infty$ we have two independent 2D atoms, both with configuration (0,5) [3]. Fig. 3(a) shows the eigenfrequencies as a function of $d$ and Fig. 3(b) the distance of the electrons from the center of the confinement potential. From a first glance we have four transition regions, namely around $d = 0.16, d = 0.68, d = 0.78$ and $d = 0.90$. The spatial configurations of the electrons are depicted in the inset of Fig. 3(b).

**Region 1** (around $d = 0.16$) and **Region 2** (around $d = 0.68$). The transitions in both regions are qualitatively similar and therefore we limit our discussion to $d \approx 0.68$. At these transition points there is no abrupt change of the configuration, but within a small $d$-region the radii of the electrons changes appreciably, but continuously (Fig. 3(a)). Fig. 3(b) shows that this region is delimited by two eigenfrequencies which become zero and consequently these transitions are induced by the softening of a mode. In Fig. 3(c) the second derivative of the energy with respect to $d$ is given. There are two discontinuities in the second derivative, namely at $d = 0.68120$ and $d = 0.68215$ while the first derivative is continuous. In the inset of Fig. 3(c) the spatial configuration is shown at $d = 0.681$ (circles) and at $d = 0.683$ (triangles). The open and closed symbols refer to electrons belonging to different layers. Notice that no qualitative changes of the configuration occurs at the second order transitions. For region 1 we found similar second order transitions at $d = 0.15775$ and $d = 0.16175$.

**Region 3** (around $d = 0.78$). The transition occurs at $d = 0.77605$ where the position of the electrons changes discontinuously. There is a jump in the eigenfrequencies and in the first derivative of the energy. Clearly this is a first order transition which is similar to those found for the above molecule with two times three electrons.

**Region 4** (around $d = 0.90$). In this region there are two transitions. The first at $d = 0.88585$, and the second at $d = 0.90588$. Only in the latter the number of electrons on the different rings changes. It is a first order transition and the eigenfrequencies exhibit a jump. On the other hand the transition at $d = 0.88585$ does not correspond to a change in the number of rings, however there is a jump in the position of the electrons. It is a first order transition with a discontinuity in the frequency spectrum in which the smallest non zero eigenfrequency decreases for $d \to 0.88585$ but stays different from zero.

For the artificial molecule in which each atom contains six electrons we find both first and second order transitions. With increasing $d$ there are three first order transitions, namely at $d = 0.2965, d = 0.5125$ and $d = 0.5345$, followed by one second order transition induced by the softening of a mode at $d = 0.9259$. The latter is different from the previous molecule with two times five electrons where there are narrow $d$-regions delimited by two second order transitions. For $d < 0.2965$ the configuration is (2,4)/(1,5), at the first order transitions the position
of the particles changes discontinuously while at the second order transition the position of the particles changes continuously and after it the outer electrons form a perfect circle with the inner electrons sitting in the center, resulting in the configuration \((1,5)/(1,5)\).

The groundstate \(^2\) of the artificial atom with seven electrons is \((1,6)\), and with eight electrons is \((1,7)\). Therefore, we expect that the above new type of transition also takes place for the artificial molecule with two times seven and two times eight electrons. For the molecule in which both atoms contain seven electrons we find two narrow \(d\)-regions delimited by two second order transitions (the first at \(d = 0.360375\) and \(d = 0.360495\), the second at \(d = 0.504625\) and \(d = 0.504865\), a first order transition at \(d = 0.5425\), and indeed the second order transition at \(d = 0.9259\). Fig. 4 shows the frequency spectrum and the position of the particles near the last transition. At the second order transition at \(d = 0.9259\) the system transforms from an imperfect ring structure with the middle electron outside the center of the system to a perfect ring with one electron in the center of each layer (see Fig. 4(b)).

For the latter molecule another phenomenon occurs. Around \(d \approx 0.74\) the smallest non zero eigenfrequency decreases substantially but does not become zero while the other eigenfrequencies and the position of the particles changes fast but continuously with \(d\). It is clear that this is not a first, second or third order transition as is made clear in the inset of Fig. 4(b) where we plotted the second and third derivative of the energy with respect to \(d\). Therefore we refer to this as a continuous transition with a lambda-like shape of the derivative \(\partial^3 E / \partial d^3\).

In conclusion, we have presented the results of a numerical calculation of the groundstate configuration, its energy and the spectrum of the normal modes of classical double-layer atoms. There is a vertical Coulomb coupling between the electrons constituting the molecule. These artificial molecules undergo structural transitions as function of the distance between the layers. We found first and second order transitions. For first order transitions the position of the particles changes discontinuously and the eigenfrequencies exhibit a jump. For second order transitions the position of the particles changes continuously and these transitions are induced by the softening of an eigenmode. The different transitions are summarized in Fig. 4 in a \((d, N)\) diagram for molecules containing up to \(2N = 20\) electrons. Although we assumed that: 1) both atoms of the molecule contain the same number of electrons, and 2) both atoms have the same confinement potential, a rich variety of structural transitions are found. It is expected that if we relax these assumptions new configurations are possible.

The present system constitutes a simple non trivial model system exhibiting a rich variety of structural transitions. This finite system correctly describes the local properties of structural phase transitions. Furthermore, we found additionally: 1) transitions which are delimited by recurring second order transitions, and 2) a lambda type of transition without a divergence in the second derivative of the energy. It should be noted that in the limit \(N \to \infty\) we obtain the system of two parallel classical two-dimensional electron gases, each of them form a Wigner crystal which was studied in Ref. [3].

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**FIG. 1.** The energy of the groundstate (plotted as dots) and its first derivative with respect to the lateral distance \(d\) between the two atoms constituting the artificial molecule for \(2N = 6\). The inset shows the frequency of the normal modes of the groundstate. The configurations are schematically represented in the figure where solid and open dots correspond to electrons in different layers.

**FIG. 2.** (a) The eigenfrequencies of the groundstate for the artificial molecule with two times five electrons as a function of the distance \(d\) between the two atoms. (b) The distance of the different electrons from the center of the confinement potential. The three groundstate configurations are also shown in the inset of the figure.
FIG. 3. (a) The distance of the electrons from the center of the confinement potential for the artificial molecule with two times five electrons as a function of $d$. (b) The lowest non zero eigenfrequency of the groundstate. (c) The second derivative of the groundstate energy with respect to $d$. The inset shows the configuration at $d = 0.681$ (circles) and at $d = 0.683$ (triangles). Open and solid symbols refer to electrons belonging to different atoms.

FIG. 4. (a) The eigenfrequencies of the groundstate for the artificial molecule with two times seven electrons as a function of $d$. (b) The distance of the electrons from the center of the confinement potential. In the inset the second and third derivative of the energy are shown around $d \approx 0.74$.

FIG. 5. Summary of the different transitions for molecules up to $2N = 20$ electrons. ‘Continuous’ refers to a transition at which the position and the eigenfrequencies change rapidly, but continuously, with $d$ and the derivatives of the energy with respect to $d$ are continuous.
Figure 1

\[ \frac{\partial (E/N)}{\partial d} \]
Figure 2

(a) 

(b)
Figure 3

(a) Radius distribution

(b) Angular frequency vs. d

(c) Second derivative of energy per unit length vs. d
Figure 4

(a) Graph showing the relationship between $\omega$ and $d$ with curves representing various data points.

(b) Graph showing the relationship between radius and $d$ with inset graphs for $\frac{\partial^2 (E/N)}{\partial d^2}$ and $\frac{\partial^3 (E/N)}{\partial d^3}$ with curves indicating changes over the range of $d$ values.
Figure 5

- □ first order
- ○ second order
- ◆ continuous