Thymol polymeric nanoparticle synthesis and its effects on the toxicity of high glucose on OEC cells: involvement of growth factors and integrin-linked kinase

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Background: Nowadays, the drug delivery system is important in the treatment of diseases. Purpose: A polymeric nanoparticle modified by oleic acid (NPMO) as a Thymol (Thy) drug release system was synthesized from Thymbra spicata and its neurotrophic and angiogenic effects on rat’s olfactory ensheathing cells (OECs) in normal (NG) and high glucose (HG) conditions were studied.

Methods: The NPMO was characterized by using different spectroscopy methods, such as infrared, HNMR, CNMR, gel permeation chromatography, dynamic light scattering, and atomic force microscopy. Load and releasing were investigated by HPLC. The toxicity against OECs diet-induced by MTT assay. ROS and generation of nitric oxide (NO) were evaluated using dichloro-dihydro-fluorescein and Griess method, respectively. The expression of protein integrin-linked kinase (ILK), vascular endothelial growth factor (VEGF), brain-derived neurotrophic factor (BDNF), and nerve growth factor (NGF) were evaluated by Western blotting.

Results: ThyNPMO is desirable for transferring drug as a carrier. The amount of Thy and extract (E) loaded on NPMO estimated at 43±2.5% and 41±1.8%, respectively. Then, 65% and 63% of the drug load were released, respectively. Thy, ThyNPMO, E, and ENPMO prevented HG-induced OECs cell death (EC50 33±1.5, 22±0.9, 35±1.8, and 25±1.1 μM, respectively). Incubation with Thy, ThyNPMO, E, and ENPMO at high concentrations increased cell death with LC50 105±3.5, 82±2.8, 109±4.3, and 86±3.4 μM, respectively in HG states.

Conclusion: OECs were protected by ThyNPMO and ENPMO in protective concentrations by reducing the amount of ROS and NO, maintaining ILK, reducing VEGF, and increasing BDNF and NGF. The mentioned mechanisms were totally reversed at high concentrations.

Keywords: thymol, polymeric nanoparticle, olfactory ensheathing cell, drug delivery, high glucose

Introduction

Nanoscale drug carriers are generally synthesized from natural and polymeric materials for slow and targeted release in sizes of 10–1000 nm. Therefore, because the structure of drug carriers has important benefits such as biocompatibility and biodegradability, it is widely used to develop pharmaceutical industries. Nanoparticles (NP), either encapsulated or absorbed, carry the drug more effectively to be used in the target tissue. Today, anticancer effects are one of the most commonly used NPs in the treatment of diseases.
Diabetes is one of the factors that can cause micro- and macrovascular disorders that increase mortality, lower quality of life, and increase the cost of treatment in the community. Increasing the cost of this disease returns to Dove Despite the protection of Thymol on high-fat diet-induced diabetic mice have been reported. Thymol is bonded to a variety of NPs for various diseases. In this study, in addition to synthesizing and producing a polymeric NP and binding it with the active component of Thymbra spicata, Thymol, the effects and mechanisms of this drug delivery system in olfactory ensheathing cells (OECs) under HG and normal glucose (NG) were studied.

Materials and methods
Preparation of polymeric nanoparticle modified by oleic acid (NPMO)
By using CA (Carlo Erba) as AB3 monomer and glycerol (G) (Carlo Erba) as A3 monomer at different CA/G molar feed ratios, nano branching polymers (NP) were synthesized through the melt polycondensation. The mixture was cooled at room temperature. Viscose compound was dissolved in tetrahydrofuran (THF) (Merck) and filtered to obtain clear solution. The resulting solution was then concentrated under reduced pressure and was precipitated several times in cyclohexane solution (Merck). Precipitated compound was dissolved in THF and then placed in a dialysis bag (Mn cutoff 2000, Sigma-Aldrich) immersed in THF for 4 hrs and THF in the outside medium of dialysis bag was replaced with the fresh solvent. Finally, to obtain pure product as colorless and viscose compound, THF was evaporated under the reduced pressure. In the second step, NPMO was synthesized. First, 3 g NP was poured into a polymerization capsule and then oleic acid was added. The resulting mixture was then heated at 90°C, 100°C, 120°C, 140°C, and 160°C, respectively, and stirred at 1000 rpm magnetic stirrer for 5 hrs under the vacuum condition. To remove residue oleic acid, the mixture was washed several times with n-hexane. Then, it was washed in 20 mL ethanol. The produced polymer and ethanol solvent were put in a dialysis bag at room temperature for 4 hrs.

FTIR spectroscopy
The IR spectra of the NPMO was performed with a Nicolet 320 spectrophotometer FTIR which was prepared by mixing the fine powder with KBr and pressing. The spectra were obtained at a resolution of 4 cm$^{-1}$ in the range 4000–500 cm$^{-1}$.29
Nuclear magnetic resonance (NMR)
All NMR experiments were conducted on a Bruker DRX 400 (400 MHz) apparatus in D$_2$O as solvent. Identical spectra were obtained by dissolving samples in D$_2$O and the spectra were recorded at 500 MHz (in $^1$H and $^{13}$C NMR spectra for all temperatures and concentrations). The resulting data were processed and analyzed using ACDLABS/1D NMR software.

Gel permeation chromatography (GPC)
Molecular weights and distribution of the obtained NPMO were determined by means of Knauer GPC equipped with Smartline Pump 1000 with a PL Aqua gel-OH mixed-H 8 µm column connected to a differential refractometer, with water as the mobile phase at 25°C.

Dynamic light scattering (DLS)
DLS data were collected on Malvern Instruments Ltd., UK. The hydrodynamic diameters of NPMO in water were measured three times (5 run to each measurement) at 90°C to the incident beam. The reported values are number distribution intensities. The measurements were performed using the samples prepared by dispersing NPMO in 1 mM NaCl at 25°C at a ratio of 0.01%, w/v. The mean size was accounted as the average of six measurements.

Atomic force microscopy (AFM)
Using a Nanoscope IIIa Multimode scanning probe microscope (Ara-research Inc. Iran) for AFM, the morphology of the NPMO was determined. A droplet of the NPMO suspension was drying (freeze dryer) (Christ, Germany) onto a clean mica surface prior to AFM imaging. In tapping mode, images were scanned using silicon cantilevers (NSC15/AIBS) delivered by Micro Mash (Tallinn, Estonia), with a frequency around 300–330 kHz. The size of the images was 5×5 µm. The images were scanned on at least six different areas of the sample.

Plant extraction and identification of Thymol by HPLC
*Thyme spicata* aerial parts were collected from around Ilam, Iran in May 2018 during the flowering season. The identity of this plant was authenticated by the voucher specimens (NO 596) deposited in the Department of Horticulture, Faculty of Agriculture, Ilam University. After drying, the specimens were powdered and 20 g was used for extraction. Initially, the powder extracted by a Soxhlet extraction method in a water-ethanol solvent. The solvents in the extract were removed by rotary (IKA HB 10, Germany) device. The yield of extraction was 6.94% and then the extract was lyophilized and kept stored at −20°C. The lyophilized samples were dissolved in methanol and filtered through a 0.22-µm syringe filter.

HPLC method was done according to the reported procedure. A reversed-phase HPLC (Smart line; Knauer, Germany) with an ultraviolet detector (Well chrome, K-2600; Knauer) and a C18 column (Nucleosil H.P.; 25 cm×0.46 cm internal diameter, 100 Å pore size, particle size 3 µm, Knauer) using gradient elution with a UV absorbance detection was developed and validated for the determination of Thymol. Column temperature, mobile phase (0.1% formic acid in water [B] was maintained at the range from 5% to 70% and solvent methanol [A]), flow rate, injection volume, and detection wavelength were set at 25°C, 1 mL/min, 1 µL, and 284 nm, respectively. In a similar condition, Thymol standard solution (dissolved in methanol) was run. A quantity of 250 mg of dried extract was dissolved in 10 mL HPLC-grade methanol, sonicated for 15 mins, filtered and further diluted to 5 mg/mL. The peaks obtained from the *Thymbra spicata* extract were compared with Thymol standard. A stock solution of Thymol standard was prepared at 0.1 mg/mL in HPLC-grade methanol, filtered and further diluted in the same solvent to obtain 15.6, 31.25, 62.5, 125, 250, and 500 µM.

Encapsulation of Thymol and extract by NPMO
One-tenth gram (1.67×10$^{-2}$ mM) of NPMO was dissolved in 5 mL distilled water and stirred for 1 hr. Then, Thymol (Sigma Aldrich) and extract dissolved in dimethyl sulfoxide (DMSO)(Merck) as stock solutions (0.1 mM) were added dropwise to a NPMO mixture and various concentrations of Thymol and Thymol in extract were obtained (25, 50, 100, 150 µM). These solutions were sonicated at room temperature for obtaining the final product.

Load and releasing capacity
HPLC method was used to determine the loading capacity, according to previous studies. To estimate the amount of pure Thymol and Thymol in extract loaded on NPMO, the HPLC was used. The mobile phase was made up of 40% methanol and 60% aqueous solution of formic acid (0.1%). After 1 hr sonication for encapsulation, water solutions of NPMO-pure Thymol and NPMO-Thymol in extract were prepared. To
remove the non-encapsulated pure Thymol and Thymol in extract residue, the solution was centrifuged at 10,000 rpm and, after precipitation, the supernatant was filtered.\(^{39}\) An aliquot of the solution after filtration was injected into the HPLC to determine the concentration of encapsulation.

In vitro release of pure Thymol and Thymol in extract from NPMO was carried out by dissolving 5 mg of pure Thymol and Thymol in extract loaded NPMO in 3 mL of PBS (0.1 M, pH 7.4). The NPMO solutions containing the pure Thymol and Thymol in extract (1 mL) were manipulated. At the time of sampling, the release medium was replaced by a fresh buffer subjected to HPLC for analysis. Each sample was then injected into the HPLC.\(^{40}\) Limit of detection (LOD) and limit of quantitation (LOQ) were 52.09 and 173.63, respectively.

### Primary culturing OECs from the olfactory mucosa

Olfactory mucosa lamina propria was used in 7-day-old rat’s pups for isolation high proliferation cells. A mixture of ketamine/xylazine (60/6 mg/kg) was used for anesthesia. Then, the lower jaw and hard palate of the rat was removed and the nasal septum was separated with entry to the area of the nose. With fresh PBS containing 1% of antibiotics, it was washed several times and then continued washing without antibiotics. About a third of the posterior septum that contains the olfactory epithelium and lamina propria was separated and then transferred to a sterile Petri dish. For isolation of cells, 0.25% trypsin enzyme (GIBCO) was used for 10 mins and then adding FBS (GIBCO). Then, the suspension was centrifuged for 10 mins at 2000 rpm. The plate cells were suspended in 4 mL of DMEM/F-12 (GIBCO), containing FBS5% and Penicillin-streptomycin solution 1% (GIBCO) and forskolin (Sigma-Aldrich) mitogenic factor 5 \(\mu\)M and after transfer to the flask, they were stored in an incubator at 37°C and carbon dioxide 5%. The flasks were replaced 48 hrs after cultivation to remove fibroblast cells. The cells begin to stick to the bottom of the flask with two different appearances: spindle-shaped Schwann-like cells and astrocyte-like cells with a flat appearance with multiple redundancies. After 10–12 days of culture, when the bottom of flask was completely filled with proliferated cells, fourth Cell Passage was performed.\(^{41}\) The cultured cells were exposed to NG (glucose 5.5 mM), HG (glucose 30 mM), or mannitol (30 mM) at 24, 48 and 72 hrs and 48 hrs was selected.

The ethical approval for this study was obtained from the Animal Care and Ethics Committee (ACEC) of the Ilam University of Medical Science (IR.MEDILAM.REC.1396.84). According to ACEC recommendations, we tried our best to minimize research animal pain and suffering. To minimize the effects of transportation-induced physiological changes in subsequent biomedical research, it is advisable to consider two factors. According to the first factor, in the present study, it was noted that the animal transfer according to the physiological conditions in accordance with the international protocols with the least stress of the animal is carried out and according to the second factor, in general, mediators of stress response to reach the desired conditions (for example, for 24 hrs) were considered.

### MTT assay

The cell viability and toxicity of the drug delivery system of various concentrations were evaluated by the MTT method.\(^{42}\) Based on this method, the OECs (3×10^4 cells per well) were plated onto 96-well culture plates with different concentrations (5, 10, 20, 40, 80, 120, and 180 \(\mu\)M) of Thy, ThyNPMO,E, and ENPMO. The cultured cells were kept in incubator for 24, 48, and 72 hrs in 5% carbon dioxide and humidity in 37°C. Then, 10 \(\mu\)L of the MTT labeling reagent was added and then the plates were incubated for 4 hrs. The final product, formazan, was dissolved in 100 \(\mu\)L of DMSO and after being shaken at 37°C; its absorption was measured at 594 nm with an ELISA reader (Spectra MAX; Molecular Devices, USA).

### Detection of ROS

The amount of intracellular reactive oxygen species was measured by 2’, 7’-dichlorofluorescein diacetate (DCFH-DA) probe.\(^{43}\) OEC cells (3×10^4 cells/well) incubated with 10 \(\mu\)M DCFH-DA at 37°C for 30 mins and in fluorescent plate reader (Biotek-FLx800) analyzed at 485 nm excitation and 520 nm emissions. An H2O2 standard (10–200 nM) curve was used for ROS production.

### Assay of NO

By Griess reaction, the amount of generation of NO (nitrite) was assessed.\(^{44}\) After 48 hrs, the cells (3×10^4 cells/well) were treated with various concentrations (20, 40, 80, and 120) of Thy, ThyNPMO, E, and ENPMO. Then, 50 \(\mu\)L of the culture supernatant was mixed with 50 \(\mu\)L of the Griess reagent and incubated for 10 mins at room temperature. Absorbance calculated at 550 nm in a...
plate reader (Dynex MRX, USA); and sodium nitrate standard concentrations solutions were used as standard.

Investigating the expression of ILK, VEGF, BDNF, and NGF proteins by Western blot

The Western blot was explained according to what was previously tested. Treated OEC cells were homogenized under conditions. The culture cell suspension was prepared in lysis buffer. Homogeneous cell suspensions were centrifuged at 13,000× g for 20 mins at 4°C, and until used, the supernatants were stored at −80°C. Content of total protein was measured according to the Micro BCA procedure (Pierce, Rockford, IL, USA). For electrophoresis, the equal amount of protein (3000 µg) in each sample was loaded onto a polyacrylamide gel (10%). The proteins were electrotransferred to the PVDF membranes and then blocked with 5% skim milk and 0.1% Tween-20 in Tris-buffered saline at room temperature for 1 hr. The membranes were first incubated with the appropriate primary antibody and then conjugated to the secondary antibody. Using the ECL kit (Amersham Pharmacia Biotech Inc., Piscataway, NJ, USA), chemiluminescence detection of the immune complexes was performed, and the results were then quantified (Bio-Rad, USA). Protein analysis was performed with anti-human ILK (ab-76468, Priab 1/5000, Sec ab1/2000), VEGF, BDNF (SAB4300702, 1: 750, Sigma), and NGF (ab- P5498, Priab 1/500, Sigma).

Statistical analysis

Each test was repeated at least three times, with essentially identical results. Statistical analyses between different treatments or groups were determined using two-sample t-tests and ANOVA. Data were expressed as mean±SD. For all tests, P-values of <0.05 were considered significant. Terms half maximal effective concentration (EC 50) and half maximal lethal concentration (LC 50) refer to the concentration of a chemical, drug, or toxic substance that produces a response halfway between the baseline and maximum after a specified exposure time.

Results

Evaluation of hyperbranched polyester

Using spectroscopic and thermal analysis methods, including IR spectra, 1H, and 13C NMR spectra, the structure of the synthesized NPMO was evaluated. The molecular weight was determined using GPC. By using DLS, the particle size of the NPMO in water was determined. Using AFM, the morphology of the NPMO was determined.

The IR spectrum of the NP compound is shown in Figure 1. The peak appearing at 1734 cm⁻¹ is related to the carboxyl ester group resulting from the combination of glycerol and CA. The 3375 cm⁻¹ peak also indicates the abundance of hydroxyl groups in the NP compound. In the IR spectrum, the NPMO composition is also given. The peak of the carbonyl group appears at 1737 cm⁻¹. This displacement in this spectrum is due to the formation of a new carbon derived from the carboxyl group, which is derived from the combination of hydroxyl groups of the NP compound with the oleic acid. As shown in the figure, after the addition of the oleic acid group (due to the presence of a long chain of carbon) to the NP compound, the peak strength of the hydroxyl group in the NPMO composition decreased to the NP compound, with a peak of 3485 cm⁻¹.

The 1H NMR spectrum of the NPMO compound is shown in Figure 2. For a more straightforward interpretation, these combinations of hydrogens with identical positions are named with the same letters. Hydrogen A is the hydrocarbon content of the monomeric CA present in the compound, a single peak at the highest intensity at 4.37 ppm. Hydrogen B is actually a single hydrogen-related monomer glycerol, which appears at 3.87 ppm. Hydrogen C is a methylated glycerol, which has two separate couriers due to the presence of neighbor hydrogen. Hydrogen D is related to the methane adjacent to the carbonyl group in the oleic acid monomer, split into a triplet peak at 2.45 ppm due to the presence of two hydrogen neighbors. Hydrogen E is a methyl group adjacent to hydrogen D, which has been split into a five-pixel peak due to the presence of four hydrogen neighbors. Hydrogen, named after the letter “F”, appears in a single pixel at 2 ppm because of the same space-spatial position. Because of the dual-bonded proximity to hydrogen, I hydrogen has a different position with
neighboring hydrogen and appears as a peak in the range of 2.2–2.3 ppm. The G-hydrides appear to be in the form of a double peak at 5.26 ppm due to the presence on the double bond, and eventually, the J-hydrides belong to the end-methyl group of the oleic acid chain, which, with minimal displacement, is a multiplicative peak at 1.15 ppm appear.

The $^{13}$CNMR spectrum of the NPMO compound is presented in Figure 3. Regarding the fact that there are 12 types of carbon in the synthetic composition, in the CNMR spectrum, this combination has a 12-peak index in the carbon compounds of this composition, which, according to the form, is named all the carbons with the letters A–M, each of which are on the spectrum. Equivalent carbon is indicated by a letter as described below. Carbon A is the middle carboxylic acid group of CA, which appears at the end of the spectrum with a displacement of 177 ppm. Carbon B is the fourth type of CA that appears at 42 ppm. Methionine CA carbonated with the letter C appears at 28 ppm. The two Carboniferous carbon represented by the letter D appear at 172 ppm. The middle carbon of glycerol is marked with the letter E, which is given at 73 ppm. Methyl glycerol carbonates have been identified with the letter F, which appear as a peak at the highest intensity of 68 ppm. Carbon is the carboxyl oleic acid group with the letter G, which has the largest displacement and peak at 183 ppm. Carbonyl adjacent carboxyl oleic acid group has a specific displacement due to its direct connection with the carboxyl group, with a different displacement than its other neighboring carbon, with the letter H marked with a peak of 38 ppm. Carbon labeled with the letter “I” has a roughly identical position and appears as a peak of more intensity than the other oleic acid carbon at 26 ppm. The adjacent dual-carbon carbons are positioned differently from other neighboring carbon and are marked with the letter J and have a peak at 21 ppm. The carbons that are connected by a double bond are named with the letter K and appear as a peak at 130 ppm. At the end of the carbon, the methyl group has appeared with a minimum displacement of 9 ppm.

The GPC diagram of NPMO is shown in Figure 4. This molecular weight chart is equivalent to 4221.72 g/mol for synthetic composition. Also, the single-medal of the
obtained diagram can be expressed by the purity of the resulting sample.

By using the DLS diagram (Figure 5), the size of the NPMO is monitored in water at 25°C. The average shape shows the NPMO diameter based on intensity. The peak width of the courtyard is indicative of the fact that there is also a particle size sample solution larger than 200 nm, but the diameter of most particles is estimated to be about 200 nm (PDI = \([402.768/200]^2 = 4.05\]) according to the chart, which is a good measure for drug delivery.

The AFM results allowed morphology observation of the NPMO at two magnifications at the micro level (Figure 6). Submicrometer sizes of the NPMO are shown in Figure 6A–C. As is seen in Figure 6C, one crystal was chosen, its length and width with cross-sectional profile is shown in Figure 6. In the cross-sectional profile along the length and width of the crystal, interchange of heterogeneous wrinkles is visible.\(^\text{46}\)

**Extraction and identification**

Based on the HPLC chromatogram, the standard Thymol broad peak was obtained at a retention time at 5.333 mins at a wavelength of 284 nm. Thymol in the ethanolic extract of the *Thymbra spicata*, in contrast to standard Thymol, was in the same condition with a retention time of 5.233 mins (Figure 7A and B). Figure 7 shows that the well-separated peak of Thymol in *Thymbra spicata* extract was compared to standard Thymol, upon application of the developed method. The quantitative analysis revealed that Thymol was found to be predominant per ethanol fraction (18.64 mg/g Thymol) of *Thymbra spicata* extract. The assay method was validated and the calibration curve was linear:

**Evaluation of the load and releasing**

By HPLC, loading capacity and release of free Thy and Thy in the E were evaluated from the NPMO. The amount of Thy and E loaded on NPMO was calculated using the standard Thymol-free curve. After three repetitions and injection into HPLC, the loading rate of Thy and E was estimated at 43±2.5% and 41±1.8%, respectively.

Free Thymol and Thymol in the extract were released in two fast and slow phases of the NPMO. In the first 12 hrs, approximately 65% of loaded Thy and 62% of loaded E were released from a fast gradient of ThyNPMO and
ENPMO, and the remaining 80–85% to 48 hrs were released at a lower rate and less slope. NPMO showed a slow biphasic kinetics with a faster release phase during the initial period as shown in Figure 8.

**Figure 4** Gel permeation chromatography image of polymeric nanoparticles modified by oleic acid (NPMO).

**Figure 5** Dynamic light scattering diagram of polymeric nanoparticles modified by oleic acid (NPMO).
Thy, ThyNPMO, E, and ENPMO effects on cell viability

After incubating of OECs with NG, HG, and HM at different times (24, 48, and 72 hrs), cell viability significantly decreased in HG conditions in 48 hrs ($p<0.001$ vs NG control). Thy, ThyNPMO, E, and ENPMO at a low concentration, ie, 5, 10, 20, and 40 $\mu$M (EC50 33±1.5, 22 ±0.9, 35±1.8, and 25±1.1 $\mu$M, respectively) ($p<0.001$ vs HG control) prevented cell death from HG. However, high concentrations, ie, 80, 120, and 180 $\mu$M, led to a reduction in cell viability with LC50 values of 121±5.3, 100±4.2, 125±5.8, and 106±5.3 $\mu$M, respectively in NG and 105±3.5, 82±2.8, 109±4.3, and 86±3.4 $\mu$M, respectively in HG states in OECs. There was no evidence of cell death of Thy, ThyNPMO, E, and ENPMO at low concentrations in NG conditions, but at high concentrations, cell viability reduced in OECs ($p<0.01$ vs NG, $p<0.001$ vs HG) (Figure 9). Molecular-level experiments were limited to 20 and 40 $\mu$M as low and 80 and 120 $\mu$M as high Thy, ThyNPMO, E, and ENPMO concentrations which showed cytoprotective and cytotoxic activities, respectively.

Effect of Thy, ThyNPMO, E, and ENPMO on ROS generation

The formation of ROS in HG conditions significantly increased ($p<0.001$ vs NG). Low concentrations of Thy, ThyNPMO, E, and ENPMO have prevented the formation of the ROS from HG conditions in the OECs. However, Thy, ThyNPMO, E, and E-NPMO at high concentrations increased the production of ROS in both HG and NG conditions. Mannitol did not alter intracellular ROS (Figure 10).
The production and release of NO in HG conditions increased compared to NG (p<0.001 vs NG). Incubating OECs, with low concentrations of Thy, ThyNPMO, E, and ENPMO prevented the release and production of NO in HG conditions. However, at high concentrations, nitric oxide increased in both under NG and HG conditions. Incubation of OECs with L-NAME (1 mM) inhibited NO release under our experimental conditions (Figure 11).

**Effect of Thy, ThyNPMO, E, and ENPMO on ILK protein expression**

The HG condition in the OECs for 48 hrs increased the ILK protein expression. This situation was also seen in mannitol (Figure 12). Low concentrations of Thy and E maintained ILK overexpression caused by HG whereas at high concentrations decreased ILK expression in both NG (p<0.01–0.001 vs NG control) and HG states (p<0.001 vs HG control).

**Effect of Thy, ThyNPMO, E, and ENPMO on VEGF protein expression**

The HG in the OECs increased the VEGF protein expression (p<0.001 vs NG) whereas mannitol did not affect the expression of VEGF. Treatment with low concentrations of Thy, ThyNPMO, E, and ENPMO decreased VEGF protein expression under HG condition (p<0.001 vs NG), though treatment with high concentrations caused an increase in VEGF protein expression in both NG and HG states (Figure 13).

**Effect of Thy, ThyNPMO, E, and ENPMO on BDNF protein expression**

The BDNF protein expression decreased in OECs in HG state (p<0.001 vs NG) whereas mannitol did not affect. Treatment with low concentrations of Thy, ThyNPMO, E,
and ENPMO caused an increase in BDNF expression under HG state (p<0.001 vs NG), whereas treatment with Thy, ThyNPMO, E, and ENPMO on high concentrations caused a decrease in BDNF expression in both NG and HG states (Figure 14).

Effect of Thy, ThyNPMO, E, and ENPMO on NGF protein expression
The NGF protein expression decreased in OECs under HG state (p<0.001 vs NG) whereas mannitol did not affect. Treatment with low concentrations of Thy, ThyNPMO, E, and ENPMO caused an increase in NGF expression under HG state (p<0.001 vs NG), whereas treatment with Thy, ThyNPMO, E, and ENPMO on high concentrations caused a decrease in NGF expression in both NG and HG states (Figure 15).

Discussion
To reduce the side effects and reduction of dosing intervals, researchers are now focusing on drug delivery and design systems. In this regard, some components of medicinal plants such as Thymol, with a terpenoids structure, are used for a variety of disorders and diseases. However, the purpose of this study was to synthesize and evaluate the neurotrophic effects of Thy, ThyNPMO, E, and ENPMO in OECs in normal and HG (glucotoxicity) and their possible mechanisms.

Various NPs, such as polymeric, ceramic, metallic based ones, micelles, liposomes, carbon nanotubes, and dendrimers have been used for a variety of research. Recent research has focused on the possibility of using hyperbranched and functionalized polymers to enhance the efficiency of drug delivery systems due to their specific characteristics, such as low viscosity, high solubility, plenty of terminal groups, and lack of chain entanglements. Hyperbranched polymers have a similar configuration to dendrimers in terms of apparent structure, but they are more easily synthesized and functionalized. In addition, to arrange their solubility, compatibility, reactivity, adhesibility to specific surfaces, self-assimilability, and chemical recognizability, end-groups of hyperbranched polymers can be easily modified. It has also been documented
that hydrophilic poly (ethylene glycol) capped poly (lactic-co-glycolic) acid NPs play an important role in glucose balance through the release of insulin-loaded.\textsuperscript{55} According to our findings, NPMO produced by thermal polycondensation has a good ability to carry material due to its biocompatibility.\textsuperscript{37} Due to factors such as small size and excellent biocompatibility, NPMOs are easily fluid in the bloodstream. Therefore, they will reach target tissues as well as increase the likelihood of binding to cell receptors.\textsuperscript{56}

In this study, NPMO was synthesized with high water solubility and biocompatibility that use as promising materials for biomedical applications. This NPMO has some cavities for loading drugs and it was used for loading Thy and extract as a diabetes drug release system. It seems that all interactions between Thymol and NPMO were noncovalent. It has been investigated that NPMO can penetrate the cell through various mechanisms, including direct diffusion from the plasmid membrane or through receptors.\textsuperscript{57,58} In an investigation of NPs, citric glycerol hyperbranched polyester was synthesized in different concentrations that were monitored cisplatin as an anticancer drug for loading in polyester.\textsuperscript{20}

In this study, different analytical techniques have been used to get NPMO structure, including FTIR, \textsuperscript{1}HNMR and \textsuperscript{13}CNMR, DLS, GPC, and AFM. The FTIR spectrum demonstrated in polymeric NP, carbonyl ester group at 1734 cm\(^{-1}\), hydroxyl groups at 3375 cm\(^{-1}\), and for NPMO carbon derived from the carbonyl group at 1737 cm\(^{-1}\) and hydroxyl group at 3485 cm\(^{-1}\). The \textsuperscript{1}HNMR spectrum showed that there are seven types of hydrogens. Hydrogen A is at 4.37 ppm. Hydrogen B is at 3.87 ppm. Hydrogen C is a methylated glycerol, which has two separate couriers due to the presence of neighbor hydrogen. Hydrogen D is at 2.45 ppm due to the presence of two hydrogen neighbors. Hydrogen E is a methyl group adjacent to hydrogen D. Hydrogen F appears in a single pixel at 2 ppm because of the same space-spatial position. Hydrogen I has a peak in the range of 2.2–2.3 ppm. The G-hydrides appear to be in the form of a double peak at 5.26 ppm. Hydrogen J is a multiplicative peak at 1.15 ppm appear. The \textsuperscript{13}CNMR spectrum of the NPMO compound displays 12 types of carbon in the synthetic composition. Carbon A is displacement of 177 ppm and carbon B is at

![Figure 10](https://www.dovepress.com/)

**Figure 10** Effect of Thy, ThyNPMO, E, and ENPMO on ROS generation in NG and HG states. Bar graphs showing ROS level (%) in olfactory ensheathing cells (OECs) exposed to normal glucose (NG), high glucose (HG), high mannitol (HM), NG plus Thymol (NG-Thy), HG plus Thymol (HG-Thy), NG plus Thymol polymeric nanoparticles modified by oleic acid (NG-ThyNPMO), HG plus Thymol polymeric nanoparticles modified by oleic acid (HG-ThyNPMO), NG plus extract (NG-E), HG plus extract (HG-E), NG plus extract polymeric nanoparticles modified by oleic acid (NG-ENPMO), HG plus extract polymeric nanoparticles modified by oleic acid (HG-ENPMO). Data are expressed as mean±SEM. *p<0.01 vs NG and HM, **p<0.001 vs NG and HM, ### p<0.001 vs HG. $\psi$ p<0.001 thymol vs ThyNPMO, \#\# p<0.001 E vs ENPMO.
42 ppm. Methionine CA carbonated with the letter C appears at 28 ppm and carbon D appears at 172 ppm. Carbon E is given at 73 ppm and F, which appears as a peak at the highest intensity of 68 ppm. Carbon G has the largest displacement and peak at 183 ppm. Carbon H marked with a peak of 38 ppm and I has a roughly identical position at 26 ppm. The adjacent dual-carbon carbons J have a peak at 21 ppm and K appears as a peak at 130 ppm. At the end of the carbon, the methyl group has appeared with a minimum displacement of 9 ppm. The obtained molecular weight (GPC technique) for copolymer was around 4221.72 g/mol. According to the DLS tests, hydrodynamic diameter of hyperbranched polyester is around 200 nm. With regard to the results of AFM, it can be concluded that NPMO is a natural micro- and mesoporous material with polymodal pore size distribution. The microporosity is connected with the NPMO, while the mesoporosity is caused by cleavage phenomenon of the NPMO.

By HPLC method, it is possible to evaluate the amount of NPMO loading capacity. In this analysis, subsequent to drawing standard curve amount of the loading rate of Thy and Extract was estimated at 43±2.5% and 41±1.8%, respectively. In one study, chitosan NP of thymol with anti-bacterial effects had a loading capacity of about 2.5%. However, in another study, Thymol nanospheres as an effective anti-bacterial agent, similar to that of our study, was reported to be about 43%.

It is important to note that the release pattern of drug from NPs and the duration of exposure to cells, as a drug release system, can have higher efficacy and safety. In our study, approximately 65% of Thymol and 62% of extract were released from the NPMO in the first 12 hrs (fast phase) and the remaining 80–85% in the next 48 hrs was released with slower kinetics and less slope, which indicated two phases the release of Thymol from NPMO. There are several mechanisms, including surface erosion, disintegration, diffusion and desorption for drug or essential oil release from NP and microparticles. The initial and rapid release of the drug from the polymer is more closely related to the particles of the drug that are surface-coated with polymeric NPs, and thus the release of the drug increases. Due to the drug absorbed on the NPMO’s surface, Thymol was already quantified in the first hours of experiment and it also called the burst effect. In a study, it has been shown that Thymol has released about 60% of NPs from natural lipids for up to 18 hrs. In another study, it has been shown that Thymol loaded with hierarchically structured biogenic silica in the first 48 hrs has been released about 50% of
Numerous studies have reported that increased chronic blood glucose not only causes glucose metabolic disorders in the brain but also destroys the nervous system and impairment cognitive activity. Disease related to diabetes includes Alzheimer’s disease, dementia, degradation of the peripheral nervous system, and neuronal disorders of the eye. Hyperglycemia by increasing the ROS causes neuropathy due to oxidative stress. Moreover, angiogenesis activity has also been shown to associate with an over-expression of a PI3 kinase–Akt–eNOS pathway and an increased VEGF production.

In the present study, treatment with Thy, ThyNPMO, E, and ENPMO at low concentrations (5, 10, 20, and 40 µM) prevented HG-mediated cell death with EC50 values of 33 ±1.5, 22±0.9, 35±1.8, and 25±1.1 µM, respectively and at high concentrations (80, 120, and 180 µM) reduced cell viability with LC50 values of 121±5.3, 100±4.2 µM, 125±5.8, and 106 ±5.3, respectively in NG and 105±3.5, 82±2.8, 109±4.3, and 86±3.4 µM, respectively in HG states in OECs. It seems that cytotoxicity of Thy, ThyNPMO, E, and E-NPMO is almost equipotent in either NG or HG conditions. It has already been observed that Thymol with EC50=0.74 µM has been used by inhibiting acetylcholine esterase in Alzheimer’s disorders. Also, Thymol is used by EC50=12 µM, due to its effect on cortical cells and the mechanism of GABA stimulation, for neurodegenerative disorders. But regarding toxic effects, it has been reported that Thymol triggered cytotoxicity inMCF-7 breast cancer cell lines with an LC50 of 2.5 mg/mL. Moreover, it has been shown that the Ocimum kilimandsc hariicum Leaf Extract Engineered Silver Nanoparticles which contains terpenoids-like Thymol structures reduce the cell viability of the Aedes aegypti with LC50=0.009 ppm.
It has been investigated that HG (35 mM) induces hyperglycemic stress, ROS production and NO generation, injury of neuronal cells. Oxidative stress through three mechanisms effect on energy level and survival in the mitochondria of neuronal cells. First, in the presence of excess $O_2^-$ NO is converted to ONOO$^-$ and in the following, ONOO$^-$ profoundly affects mitochondrial function and inhibits ATP synthesis. Second, mitochondrial oxidative stress through excess $O_2^-$ and ONOO$^-$ production inhibits the import of essential proteins to the mitochondria. Finally, oxidative damage of existing inner membrane proteins induces membrane permeability transition, a permeabilization of the mitochondrial inner membrane that causes apoptosis in cells.

In the present study, HG significantly increased intracellular ROS formation. Incubation of the cells with low concentrations of Thy, ThyNPMO, E, and ENPMO and vitamin C (100 mM) prevented this increase in fluorescence. However, high concentrations of Thy, ThyNPMO, E, and ENPMO increased ROS generation in both NG and HG states. NO release was significantly reduced in HG-cultured OECs. Treatment with low concentrations of Thy, ThyNPMO, E, and ENPMO restored NO release in OECs exposed to HG; however, NO release was reduced after treatment with high concentrations of Thy, ThyNPMO, E, and ENPMO in both NG- and HG-treated OECs. It has been shown that terpenoids, such as Thymol and carvacrol, have inhibitory effects on the central nervous system by inhibiting the release of NO in microglial cells. In addition, to the effects in NO, it has also been effective on intracellular reactive oxygen species in microglial cells in vitro.

With the effect of *Vernonia amygdalina* that contains the structure of terpenoids on the cells of the rat brain tissue, it has been shown that in conditions of HG uptake in these cells, the amount of ROS increased, the amount of glutathione decreased, and thus the amount of NO also increased.

**Figure 13** Effects of Thy, ThyNPMO, E, and ENPMO on VEGF in OECs cell in normal glucose (NG) and high glucose (HG) states. Protein expression of VEGF (%) in olfactory ensheathing cells (OECs) exposed to normal glucose (NG), high glucose (HG), high mannitol (HM), NG plus Thymol (NG-Thy), HG plus Thymol (HG-Thy), NG plus Thymol polymeric nanoparticles modified by oleic acid (NG-ThyNPMO), HG plus Thymol polymeric nanoparticles modified by oleic acid (HG-ThyNPMO), NG plus extract (NG-E), HG plus extract (HG-E), NG plus extract polymeric nanoparticles modified by oleic acid (NG-ENPMO), HG plus extract polymeric nanoparticles modified by oleic acid (HG-ENPMO). Data are expressed as mean±SEM. *p<0.01 vs NG and HM, **p<0.001 vs NG and HM, ## p<0.001 vs HG. $p<0.001$ thymol vs ThyNPMO, ## p<0.001 E vs ENPMO.
Therefore, increased antioxidant availability may be helpful in preventing or slowing the progress of various oxidative stress-related diseases.

Endothelial ILK plays an important role in vascular endothelial cell growth. ILK expression has been reported to be rapidly controlled in diabetes, and for this reason, some authors find it effective in pathogenesis of diabetes mellitus.

It has been reported that knocking down of ILK gene expression with siRNA inhibited the elevation of VEGF and ICAM 1. Also, it has been suggested that ILK has been involved in the response of cells to HG and may, therefore, play a role in the pathogenesis of diabetic retinopathy. It has been reported that terpenoid is a potent activator of Protein kinase C (PKC), and on the other hand, the synaptogenesis of astrocyte cells increases with integrin that acts through PKC. Thus, terpenoids may increase intragranularity. In the present study, we demonstrated that in the OECs, the HG condition increased the ILK protein expression. This situation was also seen in mannitol. Thy and E, at low concentrations maintained ILK overexpression caused by HG whereas at high concentrations decreased ILK expression in both NG and HG states. In the following, our results showed that HG in the OECs increased the VEGF protein expression whereas mannitol did not affect. Treatment with low concentrations of Thy, ThyNPMO, E, and ENPMO decreased VEGF protein expression under HG state, though treatment with high concentrations caused an increase in VEGF protein expression in both NG and HG states. As well as, hyperglycemia effects can produce a lot of biological effects, such as increased expression of VEGF which is very crucial for the development of diabetic retinopathy. In a study using Thymol at 40 mg/kg per day for 5 weeks, the activation of VEGF was significantly inhibited. Part of Thymol’s protective

![Figure 14](https://example.com/figure14.png)

**Figure 14** Effect of Thy, ThyNPMO, E, and ENPMO on BDNF in OECs in normal glucose (NG) and high glucose (HG) states. Protein expression of BDNF (%) in olfactory ensheathing cells (OECs) exposed to normal glucose (NG), high glucose (HG), high mannitol (HM), normal glucose (NG), high glucose (HG), high mannitol (HM), NG plus Thymol (NG-Thy), HG plus Thymol (HG-Thy), NG plus Thymol polymeric nanoparticles modified by oleic acid (NG-ThyNPMO), HG plus Thymol polymeric nanoparticles modified by oleic acid (HG-ThyNPMO), NG plus extract (NG-E), HG plus extract (HG-E), NG plus extract polymeric nanoparticles modified by oleic acid (NG-ENPMO), HG plus extract polymeric nanoparticles modified by oleic acid (HG-ENPMO). Data are expressed as mean±SEM. *p<0.01 vs NG and HM, **p<0.001 vs NG and HM, # # p<0.001 vs HG, $$ p<0.001 thymol vs ThyNPMO, ¥¥ p<0.001 E vs ENPMO.
effect appears to be reduced to the expression of this protein.

It has been documented that HG level enhanced the protein expression of NGF and BDNF in Schwann cells in vitro. In the present study, we demonstrated that in OECs in HG state the BDNF and NGF protein expression decreased whereas mannitol did not affect. Treatment with low concentrations of Thy, ThyNPMO, E, and ENPMO caused an increase in BDNF and NGF expression under HG state, whereas treatment with Thy, ThyNPMO, E and ENPMO on high concentrations caused a decrease in BDNF expression in both NG and HG states. Similar to the protective status of this study, iridoid glycosides as a monoterpene agent was able to prevent the loss of the number of hippocampal cells by increasing the production of BDNF and NGF, as well as improving the memory status of the rat. Furthermore, the *Curcuma longa* L. (Curcumin 50, 100, or 200 mg/kg) containing the terpenoid structures has also been used to increase the expression of BDNF protein in the treatment of depression. It was recently found that cannabis extract, which is rich in terpenes and flavonoids, can increase the expression of NGF by enhancing the growth of the nerve in a rat model.

**Abbreviation list**

HG, high glucose; NG, normal glucose; HM, high mannitol; Thy, Thymol; NPMO, polymeric nanoparticles modified by oleic acid; E, extract; OEC, olfactory ensheathing cells; G, glycerol; CA, citric acid; NMR, nuclear magnetic resonance; FTIR, Fourier-transform infrared; GPC, gel permeation chromatography; AFM, atomic force microscopy; DLS, dynamic light scattering; NO, nitric oxide; ILK, integrin-linked kinase; VEGF, vascular endothelial...
growth factor; BDNF, brain-derived neurotrophic factor; NGF, nerve growth factor.

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**Disclosure**

The authors report no conflicts of interest in this work.

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