X-Ray-induced Deuterium Enrichment of N-rich Organics in Protoplanetary Disks: An Experimental Investigation Using Synchrotron Light

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Abstract

The deuterium enrichment of organics in the interstellar medium, protoplanetary disks, and meteorites has been proposed to be the result of ionizing radiation. The goal of this study is to simulate and quantify the effects of soft X-rays (0.1–2 keV), an important component of stellar radiation fields illuminating protoplanetary disks, on the refractory organics present in the disks. We prepared tholins, nitrogen-rich organic analogs to solids found in many astrophysical environments, e.g., Titan’s atmosphere, cometary surfaces, and protoplanetary disks, via plasma deposition. Controlled irradiation experiments with soft X-rays at 0.5 and 1.3 keV were performed at the SEXTANTS beamline of the SOLEIL synchrotron, and were immediately followed by ex-situ infrared, Raman, and isotopic diagnostics. Infrared spectroscopy revealed the preferential loss of singly bonded groups at ~1250–1300 cm⁻¹. Raman analysis revealed that, while the length of polyaromatic units is only slightly modified, the introduction of defects leads to structural amorphization. Finally, tholins were measured via secondary ion mass spectrometry to quantify the D, H, and C elemental abundances in the irradiated versus non-irradiated areas. Isotopic analysis revealed that significant D-enrichment is induced by X-ray irradiation. Our results are compared to previous experimental studies involving the thermal degradation and electron irradiation of organics. The penetration depth of soft X-rays in μm-sized tholins leads to volume rather than surface modifications: lower-energy X-rays (0.5 keV) induce a larger D-enrichment than 1.3 keV X-rays, reaching a plateau for doses larger than 5 × 10¹⁷ eV cm⁻². Synchrotron fluences fall within the expected soft X-ray fluences in protoplanetary disks, and thus provide evidence of a new non-thermal pathway to deuterium fractionation of organic matter.

Key words: dust, extinction – methods: laboratory: solid state – protoplanetary disks – techniques: spectroscopic – X-rays: stars

1. Introduction

The study of organic molecules, dust, and ices is essential for understanding the origin and evolution of the solar system. The deuterium-to-hydrogen ratio (D/H) is used as a proxy of the surrounding physico-chemical conditions affecting organics, omnipresent in inter- and circumstellar media (ISM/CSM), and occasionally delivered to Earth and other planets in the form of meteorites. Kerridge (1983) proposed early on that deuterium enrichment in meteorites had an interstellar origin. Yet, interstellar clouds can further evolve, reaching a critical mass and density, collapsing, and forming protoplanetary disks (Williams & Cieza 2011) on par with our own protosolar nebula (PSN). Isotopic fractionation can occur at any of these stages via multiple chemical pathways such as isotopic exchange reactions in the ISM (Roueff et al. 2015) and gas-surface reactions in protoplanetary disks (Henning & Semenov 2013).

Deuterated molecules have been detected in all stages of star formation; for a review, see Ceccarelli et al. (2014). In molecular clouds, detected deuterated species include DCO⁺ and DNC (van der Tak et al. 2009), H₂D⁺, and HD₂⁺ (Parise et al. 2011) among many others. In protostellar cores, detections include DCN and D₂CO (Ceccarelli et al. 2002; Roberts et al. 2002), CH₃CDD (Markwick et al. 2005), CH₂OD (Parise et al. 2006), NH₂D₀, and DNCO (Coutens et al. 2016). In protoplanetary disks such as TW Hydrae, the most well-studied protoplanetary disk because of its proximity and near face-on orientation, detections include DCO⁺ (van Dishoeck et al. 2003; Qi et al. 2008), H₂D⁺ (Ceccarelli et al. 2004), and DCN (Qi et al. 2008; Oberg et al. 2012). In the young protoplanetary disk surrounding the solar-type protostar DM Tau, detections of H₂D⁺ (Ceccarelli et al. 2004) and DCN (Qi et al. 2008) were reported but later contested by Guilloteau et al. (2006), along with the confirmed detection of DCO⁺. Chemical models including a vast network of deuterated species have pointed at the PSN origin of water found in comets and asteroids (Albertsson et al. 2014), although modeling work by Cleeves et al. (2014) argued that ion-driven deuterium pathways in the disk are inefficient, implying an interstellar origin for water. Therefore, the
dominance of deuterium fractionation at a particular stage in stellar evolution persists as an open question in astrophysics.

In the solar system, deuterated species have been measured in comets and meteorites. Recent in situ measurements in the Jupiter family comet 67P/Churyumov–Gerasimenko by the ROSETTA spacecraft found the D/H in water to be approximately three times the terrestrial value (Altwegg et al. 2015), suggesting an asteroidal rather than cometary origin for the Earth oceans and atmosphere. This contrasts with reports on the D/H ratio of the Jupiter family comet 103P/Hartley 2 (Hartogh et al. 2011), compatible with the value of Earth ocean-like water. These measurements point at mixing processes in the early PSN. Analyses of meteorites have shown deuterium enrichment of their insoluble organic matter compared to the solar hydrogen abundance (Robert & Epstein 1982; Yang & Epstein 1983), and suggested an interstellar heritage (Sandford et al. 2001; Busemann et al. 2006). D-enrichment has also been measured in interplanetary dust particles (Messinger 2000) and more recently in UltraCarbonaceous Antarctic Micrometeorites (UCAMMs; Duprat et al. 2010). However, isotopic anomalies in meteorites and comets are not necessarily fingerprints of an interstellar origin: organic matter can be further modified in the coldest outer regions of protoplanetary disks (Remusat et al. 2006; Aléon 2010), e.g., via isotopic gas-grain exchange reactions (Remusat et al. 2009).

Laboratory experiments have the potential of unveiling the chemical and physical mechanisms at the origin of isotopic fractionation. Thermal processing experiments can impact the chemical and structural properties of astrophysical dust and ice analogs, e.g., (Allamandola et al. 1988; Ehrenfreund et al. 1999; Collings et al. 2004). However, non-thermal irradiation processes are just as important in astrophysical environments where elevated dust or gas densities inhibit thermal annealing. Ionizing radiation includes charged particles (electrons or protons), ions, but also UV photons and X-rays (Feigelson et al. 2002). Recent experiments have investigated the effects of electron irradiation on the structure and the D/H signature of kerogens (natural organic macromolecules; Le Guillou et al. 2013), synthetic polymeric organics (Laurent et al. 2014, 2015) and in analogs of protoplanetary hydrous silicate dust (Roskosz et al. 2016). These experiments have revealed that non-thermal irradiation can offer viable pathways to deutrium enrichment in the solid phase.

X-rays from T-Tauri stars are associated with the energy dissipation of magnetic coronal fields (Preibisch et al. 2005). Together with UV photons, these can irradiate the disk surface, midplane and outer edges promoting photochemistry and radiolysis (Glassgold et al. 1997; Henning et al. 2010; Aresu et al. 2011). As of now, no laboratory investigations have considered the possible isotopic effects of X-ray irradiation on organic matter. Yet, we know hard X-rays (>5 keV) can strongly impact the structure of silicates in protoplanetary disks (Gavilan et al. 2016). Thus, our goal is to experimentally unveil the isotopic effects of stellar soft X-rays (0.1–2 keV) on protoplanetary organic matter.

To this end, we employ nitrogen-rich amorphous organic solids as analogs to refractory organics produced from the thermal and/or photoprocessing of N-rich ices in regions beyond the snow-line of protoplanetary disks (Greenberg et al. 1995; Ciesla & Sandford 2012). To simulate the T-Tauri X-ray field, we perform controlled experiments using synchrotron radiation in the soft X-ray range. The brilliance of synchrotron X-rays is ideal for performing highly focused monochromatic studies with remarkable reliability, allowing us to simulate astrophysical fluences.

2. Experiments

2.1. Sample Preparation

Tholins, complex organic compounds rich in carbon and nitrogen used as analogs to astrophysical solid matter (Sagan & Khare 1979), were produced using the PAMPRE setup, a low pressure (0.95 mbar) radiofrequency (RF) plasma reactor located at LATMOS (Guyancourt, France; Szopa et al. 2006). A 13.56 MHz RF power source tuned at 30 W generates a capacitively coupled plasma fed by a gas mixture of N2; CH4 = 95:5, conditions chosen for optimal tholin production (Scrimm, O’Brien et al. 2012). The plasma was produced within a vertical cylindrical cage and the substrates were placed on the bottom electrode. The produced films, rich in carbon and nitrogen, i.e., N/C = 0.2, H/C ~ 0.1 (Carrasco et al. 2016), make them comparable to the organic matter in UCCAMs with N/C = 0.05–0.12 (Dartois et al. 2013) and to refractory organics that may have formed from the energetic irradiation of nitrogen-rich ices in the outer solar system.

Two types of tholins with different thicknesses were prepared. A thin film (~300 nm) on an X-ray transparent substrate was prepared for X-ray absorption spectroscopy (XAS) with the goal of obtaining absolute absorption coefficients in the soft X-ray range. Thick films (~1200 nm) were prepared for irradiation with the goal of absorbing most of the deposited X-rays within the sample volume. The thin tholin was produced after turning on the plasma for 30 minutes. It is evenly deposited on both the Si3N4 membrane and the supporting frame made of silicon (5 × 5 mm). For this sample, ellipsometry is performed on the reflective Si frame. The Si3N4 itself is 30 nm thick and quasi-transparent to X-rays (Torma et al. 2014). Thick tholin films were deposited on single-side polished silicon substrates (9 × 4 mm) after leaving the plasma on for 150 minutes.

To determine the film thickness, we used spectroscopic ellipsometry (SE). This is an optical technique where a beam of polarized light is obliquely sent toward a film deposited on a reflective surface. Light is elliptically polarized upon reflection.

Both tan(Ψ), the ratio of the incident to the reflected light amplitude, and Δ, the relative phase change for p- and s-polarized light components, are measured, as seen in Figure 1. These determine the absolute value of the complex reflectance ratio, ρ, defined as,

\[ \rho = \frac{r_p}{r_s} = \tan(\Psi) e^{i\Delta}, \]  

where \( r_p \) and \( r_s \) are the complex Fresnel reflection coefficients of p- and s-polarized light. Using \( \rho \) and an appropriate optical model we can find the \( \chi \)-minimized dielectric properties, layer (or multi-layer) film thickness, and roughness. For this study we used a rotating spectroscopic ellipsometer (M-2000V, J.A. Woollam), which allows all wavelengths to be measured simultaneously. The light source is a tungsten lamp covering the 370–1000 nm range, incident at an angle of 70° with a spot size at the sample of ~2 × 3 mm. Modeling of the ellipsometric parameters was done with the Complete-EASE™ ellipsometry software. We selected a Tauc–Lorentz model,
suitable for amorphous organic films (Jellison & Modine 1996), providing the thickness $d$, surface roughness $r$, $E_x$ (the Tauc optical gap), and the mean squared error (MSE) between the modeled and the experimental data. The modeled parameters are listed in Table 1, showing the consistent gap value (and thus homogeneity) of films prepared under the same conditions but at different durations, in agreement with previous measurements (Mahjoub et al. 2012).

### 2.1.1. X-Ray Absorption Spectroscopy

We performed XAS measurements at the COMET experimental station, an ultra-high vacuum chamber placed in the intermediate focal point of the elastic branch of the SEXTANTS beamline (Sacchi et al. 2013) of the SOLEIL synchrotron. The samples were placed on a motorized sample holder perpendicular to the X-ray beam. Four different gratings were used: 100–300 eV, 300–500 eV, 500–800 eV, and 800–1200 eV with resolution $E/\Delta E > 10^4$. For each range two samples were measured: the thin tholin on a Si$_3$N$_4$ membrane and a bare Si$_3$N$_4$ membrane. The optical depth is obtained using the Beer–Lambert law,

$$\tau(\nu) = -\ln \frac{I(\nu)}{I_0(\nu)},$$

where $I(\nu)$ is the intensity of light at a given frequency, $\nu$, transmitted through the tholin film deposited on the membrane and $I_0(\nu)$ is the light intensity through the membrane alone.

The imaginary index of refraction can be obtained via

$$k = \frac{\tau(\nu)}{4\pi d},$$

where $d$, the film thickness, is found by ellipsometry as aforementioned. Similarly, the penetration depth ($L_p$) of a photon is given by,

$$L_p(\mu m) = \frac{\lambda(\mu m)}{4\pi k}.$$

The resulting $k$ and $L_p(\mu m)$ in the soft X-ray range for the tholins is presented in Figure 2. Fluctuations in the $k$ value around 100 eV are due to Si L-edges from the Si$_3$N$_4$ substrate. In addition to the expected C K-edge (390 eV) and N K-edge (397 eV) absorptions, we also notice the O K-edge (530 eV) and the Fe L-edges (720 eV). These are attributed to pollution and sputtering of the confining plasma cage where tholins are produced. Both $k$ and $L_p$ are the required parameters to optimize the irradiation experiments on thick tholins. While 0.5 keV X-rays penetrate the tholin film up to a thickness of $\sim$625 nm (depositing most of their energy within the 1.16 $\mu$m film), 1.3 keV X-rays have a penetration depth greater than 5 $\mu$m (i.e., only a fraction of their energy will be deposited within the film; the rest will be scattered). In this paper, we report the incident dose on each experiment and not the adsorbed dose, which varies depending on the chemical nature of the material (reflected in its absorption index) and on its thickness.

### 2.2. X-Ray Irradiation

We performed the X-ray irradiation experiments at the COMET station using two monochromatic settings, chosen to

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**Table 1**

| MSE       | Thickness (nm) | Roughness (nm) | $E_x$ (eV) |
|-----------|----------------|----------------|------------|
| $3.2 \times 10^{-3}$ | 305 ± 12        | 5.1 ± 1.1      | 2.12 ± 0.08 |
| $8.2 \times 10^{-3}$ | 1139 ± 42       | 13.2 ± 2.1     | 2.10 ± 0.05 |

Figure 1. Ellipsometric parameters $\Psi$ and $\Delta$ measured in the UV-Visible (390–1000 nm) and modeled to determine the thickness of (a) a thin tholin film, and (b) a thick tholin film, used for X-ray absorption spectroscopy and irradiation.

Figure 2. Imaginary index of refraction as a function of energy and calculated penetration length, $L_p$, in $\mu$m. The energies selected for irradiation are marked in green.
evaluate the impact of irradiation across the soft X-ray region: 0.5 and 1.3 keV. The beam was right-circularly polarized for both experiments with a spot size at the sample of 80 × 50 μm, flux density $\sim 10^{17}$ photons s$^{-1}$ cm$^{-2}$ and resolution of $E/\Delta E > 10^4$ up to 1200 eV. For each energy, tholin films were irradiated at spots separated by 300 μm for different durations: 10 minutes, 30 minutes, 1 hr, 4 hr, and 10 hr.

Following irradiation, the sample surface was examined with a white-light interferometer (smartWLI), which allowed us to record 3D topographies with depth resolution $\sim$3 nm. The result is shown in Figure 3, where X-irradiated regions are manifested by volume expansion seen as elevated craters on the tholin surface (with rising edges up to 200 nm from the surface), reticulation, and darkening. The irradiated spots extend up to $\sim$200 × 200 μm$^2$, i.e., twice as large as the incident beam size, revealing the diffusion of secondary electrons during irradiation. The central spot, corresponding to the highest X-ray dose, extends even farther out than all others.

### 3. Non-destructive Diagnostics

We performed ex-situ diagnostics of the irradiated tholins via infrared and Raman microspectrometry on the 0.5 keV irradiated tholin. These diagnostics allow us to further quantify the chemical and/or structural evolution of the irradiated tholin films, revealed by optical interferometry in the previous section. Access to infrared and Raman microscopes was kindly provided by the SMIS beamline (Dumas et al. 2006) at SOLEIL.

#### 3.1. Infrared Microscopy

Infrared spectra of irradiated and non-irradiated regions were measured to quantify the chemical evolution of the irradiated films at increasing X-ray doses. We used an iN10 microscope (ThermoFisher Scientific) coupled to a FTIR spectrometer in offline mode. Areas of 100 × 100 μm$^2$ within the irradiated regions were integrated from 4000 to 1000 cm$^{-1}$. A polynomial baseline was subtracted from the infrared spectra to normalize the continuum arising from rough surface scattering and from broad ultraviolet absorptions typical of amorphous carbon materials reaching the near-infrared (Khare et al. 1984; Bernard et al. 2006).

Figure 4 shows the evolution of the infrared vibrational bands from the initial (non-irradiated) tholin to the increasingly irradiated regions. This is detailed in Figure 5, where infrared spectra have been split into three main regions: (a) the 3600–2500 cm$^{-1}$ region, which includes stretching modes of amine groups (primary/secondary), superimposed by the methyl (CH$_3$) and methylene (CH$_2$) stretching modes; (b) the 2400–2000 cm$^{-1}$ region, which includes nitriles and isonitriles (C=≡N/N≡C) stretching modes; and (c) the 1800–1000 cm$^{-1}$ region, which includes several overlapping bands including the sp$^2$ aromatic C=C and C≡N stretching modes, sp$^3$-CH$_3$ bending modes, and sp$^3$-like C=C or C≡N stretching of defects (such as non-hexagonal rings of aromatic compounds; Carpentier et al. 2012). The fitted infrared parameters as a function of X-ray dose are listed in Table 3. For a complete attribution to vibrational modes of tholins in the mid-infrared we refer to Imanaka et al. (2004). To analyze the kinetic evolution of the main vibrational modes, we performed the deconvolution of the continuum subtracted 3.4 μm band (3100–2800 cm$^{-1}$) and on the nitrile bands (2300–2000 cm$^{-1}$), presented in Figure 6. Integrated optical depths were fitted as a function of the irradiated dose with the following exponential relation,

$$y = y_0 + Ae^{-kE_{dose}},$$

where $y$ corresponds to the integrated optical depths, $y_0$ corresponds to an asymptotic steady-state, $A$ is a constant, $k$ is the rate constant, and $E_{dose}$ is the irradiation dose.

Figure 7 shows the infrared band kinetics during X-ray irradiation. The fitted parameters are reported in Table 2. We note the rapid depletion of the C–H stretching band (2800–3000 cm$^{-1}$) by the 95% loss of its original intensity and the depletion of the N–H band and the R–N≡C at 2175 cm$^{-1}$ band both by the $\sim$67% loss of their original intensity following the maximum X-ray doses. Before the continuum baseline correction, we note that the absorbance at 4000 cm$^{-1}$ increases from 0.7 to 0.9 at the highest X-ray dose, signaling darkening of the tholin film.

We also note the growth of specific band intensities during irradiation, such as the vibrational mode at 2210 cm$^{-1}$, attributed to –CN groups attached to aromatic molecules (Khare et al. 1981; Sagan et al. 1993), and the sp$^3$ carbon defect
band between 1250 and 1300 cm\(^{-1}\), affecting the entire bending mode region between 1800 and 1000 cm\(^{-1}\). Nitriles are not completely lost but the network bearing them becomes unsaturated and more aromatic. For the longest exposures (4 and 10 hr), a new peak appearing at 2335 cm\(^{-1}\) is attributed to isocyanate R–N≡C\(=\)O modes, likely the result of

\[\text{Figure 5. Infrared spectra of irradiated tholins at 0.5 keV split into main spectral regions as defined in Section 3.1.}\]

\[\text{Figure 6. Infrared band evolution of the nitrile bands for X-ray doses at 0.5 keV. Deconvolution of the main nitrile modes for (a) the initial tholin and (b) the 10 hr irradiated tholin, show the loss of the R–N≡C band intensity for the growth of the Ar–C≡N band intensity.}\]

\[\text{Figure 7. Evolution of the intensities of the main infrared bands (normalized to the integrated optical depth of the non-irradiated tholin infrared bands) as a function of the deposited X-ray dose, displaying the favored depletion of alkyls, amines, and singly bonded isonitriles, and the growth of sp\(^3\) carbon defects. IR band kinetics are fitted with exponential functions.}\]
photo-oxidation of impurities in the tholin (noticed by the weak O edge in the XAS; Figure 2). In Figure 8, we further compare the infrared spectra of our X-irradiated tholins to a thermally annealed tholin (Bonnet et al. 2015) and to a UCAMM (Dartois et al. 2013). While the X-irradiated tholin does not present the high loss of amines that is seen in the other two, the growth of the defect band near 1300 cm⁻¹ occurs in the three cases.

3.2. Raman Microspectrometry

Raman spectra of irradiated and non-irradiated regions were measured to quantify the structural evolution (size of aromatic units, graphitization/amorphization) of the irradiated films at increasing X-ray doses. These were measured with a DXR Raman spectrometer (ThermoFisher Scientific) using a 514 nm excitation laser, with laser power ≤0.3 mW focused with a 50× objective and a spot diameter of 2–3 μm, well within the 200 × 200 μm² irradiated regions (Figure 3).

The main features in the Raman spectra of carbon-rich organic materials are the so-called G and D peaks, around ω(G) = 1560 and ω(D) = 1360 cm⁻¹, dominated by sp² sites, as visible excitation resonates with π states. We note that the non-irradiated tholin produces an intense fluorescence background in the Raman signal, an issue that could be avoided via photobleaching (which inherently transforms the material) or by employing an ultraviolet excitation source (Bernard et al. 2006). This was not a problem for the X-irradiated, nitrogen-depleted regions. The baseline-corrected Raman spectra of the irradiated tholins are shown in Figure 9, normalized to the intensity of the G-band. For the non-irradiated tholin, measurement of the Raman spectra was hindered by strong fluorescence. Raman spectra were deconvolved following the prescription in Ferrari & Robertson (2000), where the G-band is fitted with a Breit-Wigner-Fano profile and the D-band is fitted with a Lorentzian profile. This procedure provides Raman parameters including the position of each band, their full-width at half maximum (FWHM), and the ratio of their integrated intensities (I_D/I_G). These parameters are listed in Table 4.

As proposed in Ferrari (2007), the Raman parameters for completely disordered, amorphous carbons consisting of distorted sixfold rings are related as follows,

\[ I_D/I_G = C'L_o^2, \]

where C′(514 nm) = 0.0055, and L_o is the average correlation length of polyaromatic units in Å. For the X-ray irradiated tholins, the size of polyaromatic units remains almost unchanged, i.e., L_o increases very slightly after 30 minutes of irradiation and barely decreases from 1.27 ± 0.02 to 1.25 ± 0.02 nm after 10 hr, implying that no larger polyaromatic structures are created. As the X-ray irradiation dose increases, the intensity of the D-band decreases with respect to the G-band, while both the FWHM-G and FWHM-D increase.

The broadening and position shift are attributed to an increasing number of defects, including bond length and angle disorder at the atomic scale resulting from cross-linking due to new bond formation. This leads to the reorganization in the tholin structure manifested macroscopically as reticulation and volume expansion (Figure 3).

In Figure 10, we compare the evolution of Raman parameters of tholins undergoing X-ray irradiation (this study) to tholins that are thermally degraded (Bonnet et al. 2015). For the latter, heating leads not only to carbonization (preferential expulsion of heteroatoms) but also to the creation of polyaromatic structures, while in our case polyaromatic structures are almost unaffected by X-ray irradiation. However, X-ray irradiation may favor the creation of isolated over polyaromatic units, as signaled from the infrared analysis of the nitrile modes in Section 3.1.

4. Destructive Diagnostics: Secondary Ion Mass Spectrometry (SIMS)

We used SIMS to characterize the evolution of elemental abundances of the irradiated tholins at 0.5 and 1.3 keV and to detect possible isotopic effects. SIMS is a destructive diagnostic, as seen in Figure 11. We used the SIMS IMS-7f (CAMECA), installed at the Institut Jean Lamour (Nancy, France), a double-focusing magnetic sector SIMS that allows detection of isotopes at high mass resolution at the scale of a few microns. We used a high-energy primary ion Cs⁺ source (impact energy of 15 keV) for efficient sputtering with a beam current of 0.39 nA. The mass spectrometer was tuned to obtain a mass resolving power of m/Δm = 3000. We collected and quantified secondary ions of H⁺, D⁺, and ¹²C⁺, from the irradiated and non-irradiated tholin spots.

Before SIMS operation, the irradiated tholin films were gold-coated (thickness ~15 nm). Samples were inserted in the analysis chamber operating at 10⁻⁹ mbar. The ion beam was rastered over 50 × 50 μm², and using an aperture diaphragm, only the inner 35 × 35 μm² were sampled, to reduce the contribution of surface contamination to the H signal. The analyzed area was small enough to accurately probe the 200 × 200 μm² irradiated spots. In between each measurement of irradiated spots, non-irradiated areas on the same tholin film were measured and constituted reference measurements (Figure 11). Analyses were saved as depth profiles. Sputtering was stopped as soon as the substrate was reached. The sputtering depth is calibrated using the film thickness obtained via ellipsometry, which is described in Section 2.1. The D/H profiles for irradiated spots are reported as isotope fractionation, which is defined as

\[ \alpha_{D/H} = \frac{D/H_i}{D/H_o}, \]
where $D/H_o$ is the D/H ratio of the reference tholin measured near the irradiated area and $D/H_i$ is the ratio after a given X-ray dose. Similarly, the abundance of $H/C$ for an irradiated spot is parametrized as,

$$\alpha_{H/C} = \frac{H/C}{H/C_o},$$

(8)

where $H/C_o$ corresponds to the reference tholin and $H/C_i$ is the value after a certain irradiation dose. The irradiation dose is defined as,

$$E_{dose} = k \times F,$$

(9)

where $k$ is the absorption coefficient (Figure 2) and $F$ is the fluence obtained from the total flux integrated in time and divided by the irradiated area. The fitted SIMS parameters as a function of X-ray dose are listed in Table 5.

In Figure 12(a) we show the $\alpha_{D/H}$ profiles for the 0.5 keV irradiated tholin, which reach a maximum after 1 hr irradiation, corresponding to $\alpha_{D/H} = 1.9$. Longer exposures do not induce further D-enrichment. We also note that for shorter irradiation times (10 and 30 minutes) the deuterium enrichment decreases linearly in depth. This is evidence of the cumulative dose effect: the regions where the enrichment is largest, and hence are more affected by irradiation, are found near the surface, while toward the tholin/Substrate interface the enrichment decreases and becomes negligible. We hypothesize that the bulk of the tholin is shielding the lower regions from irradiation for short duration experiments. From Figure 2, 0.5 keV photons will have a maximum penetration depth in tholins at ~625 nm.

Figure 8. From top to bottom: comparison between infrared spectra of an initial tholin to an X-irradiated tholin, thermally annealed tholin (Bonnet et al. 2015), and a UCAMM (Dartois et al. 2013).

During irradiation, the released secondary electrons (photoelectrons or Auger electrons) will be diffused completely within the tholin volume. On the other hand, in Figure 12(b) we note the linearity of the $\alpha_{D/H}$ profiles for the tholin irradiated at 1.3 keV. At this energy the maximum penetration depth is $>5 \mu$m, i.e., for our 1.16 $\mu$m thick samples only ~20 percent of the incident energy will be absorbed. The penetration length of X-rays in a material depends on the X-ray energy and the absorption index of the material, both affecting the total energy absorbed. For our 1.1 $\mu$m tholin, lower-energy X-rays are absorbed more efficiently. This explains the lower fractionation $\alpha_{D/H} = 1.7$ for 10 hr irradiation at 1.3 keV compared to 0.5 keV, and the need for a higher dose to reach the enrichment plateau at 1.3 keV.

Whereas at 0.5 keV the maximum isotope fractionation is reached after 1 hr and does not evolve for longer irradiations, at 1.3 keV isotope fractionation continuously increases from 1 to 10 hr irradiation. In addition, the 10 minute experiment at 1.3 keV induces a negligible D-enrichment. Hence, higher-energy X-rays appear to be less efficient drivers of H-isotope fractionation.

The $\alpha_{H/C}$ profiles are reported in Figures 12(c) and (d), and are related to the evolution of the molecular structure during X-ray irradiation. For the shortest irradiation at 0.5 keV, tholins exhibit a modest H-depletion at the surface and a negligible effect near the tholin/Substrate interface: $\alpha_{H/C} = 0.8$ at the surface and increases regularly up to 1 hr near the surface. Hence, trends for $\alpha_{H/C}$ and $\alpha_{D/H}$ as a function of depth are opposite: D-enrichment is associated to H/C decrease (i.e., loss

### Table 3

| Time (s) | 0.5 keV Dose (eV cm$^{-3}$) | C-H | N-H | R-N$\equiv$C | C-defect |
|---------|-----------------------------|-----|-----|--------------|---------|
| 600     | $3.80 \times 10^{26}$       | 0.055 (±0.007) | 0.56 (±0.04) | 0.45 (±0.04) | 1.56 (±0.15) |
| 1800    | $1.14 \times 10^{27}$       | 0.036 (±0.008) | 0.37 (±0.03) | 0.41 (±0.03) | 1.76 (±0.11) |
| 3600    | $2.28 \times 10^{27}$       | 0.042 (±0.006) | 0.50 (±0.02) | 0.38 (±0.03) | 1.80 (±0.13) |
| 14400   | $9.12 \times 10^{27}$       | 0.028 (±0.008) | 0.32 (±0.04) | 0.32 (±0.02) | 2.09 (±0.13) |
| 36000   | $2.28 \times 10^{28}$       | 0.026 (±0.003) | 0.31 (±0.05) | 0.30 (±0.03) | 2.01 (±0.11) |

**Figure 8.** Integrated Infrared Mode

| Time (s) | 0.5 keV Dose (eV cm$^{-3}$) | C-H | N-H | R-N$\equiv$C | C-defect |
|---------|-----------------------------|-----|-----|--------------|---------|
| 600     | $3.80 \times 10^{26}$       | 0.055 (±0.007) | 0.56 (±0.04) | 0.45 (±0.04) | 1.56 (±0.15) |
| 1800    | $1.14 \times 10^{27}$       | 0.036 (±0.008) | 0.37 (±0.03) | 0.41 (±0.03) | 1.76 (±0.11) |
| 3600    | $2.28 \times 10^{27}$       | 0.042 (±0.006) | 0.50 (±0.02) | 0.38 (±0.03) | 1.80 (±0.13) |
| 14400   | $9.12 \times 10^{27}$       | 0.028 (±0.008) | 0.32 (±0.04) | 0.32 (±0.02) | 2.09 (±0.13) |
| 36000   | $2.28 \times 10^{28}$       | 0.026 (±0.003) | 0.31 (±0.05) | 0.30 (±0.03) | 2.01 (±0.11) |

**Figure 9.** Raman spectra of X-ray irradiated tholins as a function of X-ray dose at 0.5 keV. The strong fluorescence from the non-irradiated tholin hinders measurement using a 514 nm excitation source.

**Table 3**

| Time (s) | 0.5 keV Dose (eV cm$^{-3}$) | C-H | N-H | R-N$\equiv$C | C-defect |
|---------|-----------------------------|-----|-----|--------------|---------|
| 600     | $3.80 \times 10^{26}$       | 0.055 (±0.007) | 0.56 (±0.04) | 0.45 (±0.04) | 1.56 (±0.15) |
| 1800    | $1.14 \times 10^{27}$       | 0.036 (±0.008) | 0.37 (±0.03) | 0.41 (±0.03) | 1.76 (±0.11) |
| 3600    | $2.28 \times 10^{27}$       | 0.042 (±0.006) | 0.50 (±0.02) | 0.38 (±0.03) | 1.80 (±0.13) |
| 14400   | $9.12 \times 10^{27}$       | 0.028 (±0.008) | 0.32 (±0.04) | 0.32 (±0.02) | 2.09 (±0.13) |
| 36000   | $2.28 \times 10^{28}$       | 0.026 (±0.003) | 0.31 (±0.05) | 0.30 (±0.03) | 2.01 (±0.11) |

During irradiation, the released secondary electrons (photoelectrons or Auger electrons) will be diffused completely within the tholin volume. On the other hand, in Figure 12(b) we note the linearity of the $\alpha_{D/H}$ profiles for the tholin irradiated at 1.3 keV. At this energy the maximum penetration depth is $>5 \mu$m, i.e., for our 1.16 $\mu$m thick samples only ~20 percent of the incident energy will be absorbed. The penetration length of X-rays in a material depends on the X-ray energy and the absorption index of the material, both affecting the total energy absorbed. For our 1.1 $\mu$m tholin, lower-energy X-rays are absorbed more efficiently. This explains the lower fractionation $\alpha_{D/H} = 1.7$ for 10 hr irradiation at 1.3 keV compared to 0.5 keV, and the need for a higher dose to reach the enrichment plateau at 1.3 keV.

Whereas at 0.5 keV the maximum isotope fractionation is reached after 1 hr and does not evolve for longer irradiations, at 1.3 keV isotope fractionation continuously increases from 1 to 10 hr irradiation. In addition, the 10 minute experiment at 1.3 keV induces a negligible D-enrichment. Hence, higher-energy X-rays appear to be less efficient drivers of H-isotope fractionation.

The $\alpha_{H/C}$ profiles are reported in Figures 12(c) and (d), and are related to the evolution of the molecular structure during X-ray irradiation. For the shortest irradiation at 0.5 keV, tholins exhibit a modest H-depletion at the surface and a negligible effect near the tholin/Substrate interface: $\alpha_{H/C} = 0.8$ at the surface and increases regularly up to 1 hr near the interface. Hence, trends for $\alpha_{H/C}$ and $\alpha_{D/H}$ as a function of depth are opposite: D-enrichment is associated to H/C decrease (i.e., loss...
of H). As the dose increases, dehydrogenation of the organic skeleton increases and becomes more homogeneous across the tholin thickness. At 0.5 keV, $\alpha_{H/C}$ reaches a plateau after 1 hr, while at 1.3 keV the plateau is reached after 4 hr.

Figure 13 shows the average $\alpha_{D/H}$ versus the average $\alpha_{H/C}$ parameters, for both the 0.5 and 1.3 keV irradiated tholins. These are found from the average $\alpha_{D/H}$ and $\alpha_{H/C}$ values over the entire film thickness in Figure 12, and the scatter in this value determines the error bars. The strong correlation between these two parameters confirms that the preferential H over D loss (a reflection of their differences in zero-point energies) during X-ray irradiation is at the origin of the measured deuterium enrichment.

Figure 14 shows the evolution of the deuterium enrichment, $\alpha_{D/H}$, as a function of the total X-ray dose for both 0.5 and 1.3 keV. We note that the D-enrichment is slightly larger for 0.5 keV X-rays, which depose a larger energy per volume than 1.3 keV X-rays. We compare our results to the evolution observed for 30 keV electron irradiation of organic polymers (Laurent et al. 2015). We first remark that the deposited dose per volume is 1–3 orders of magnitude larger in our experiment than that for electron irradiation. Second, the total maximum D-enrichment (at the plateau) is about 20% larger in our study. X-ray irradiation acts as a bulk process due to the larger penetration depth (compared to electron irradiation) and volume diffusivity of secondary electrons, enhancing D-enrichment.

5. Discussion

The evolution of tholins induced by X-ray irradiation can be chemically quantified by the sequential depletion and growth of infrared bands, structurally by the analysis of Raman bands, and isotopically by the evolution of the SIMS profiles. For each of these processes, the evolution of the main parameters as a function of the dose can be modeled using first-order rate equations (Laurent et al. 2015),

$$\frac{dy}{dE} = Ae^{-kE_{dose}},$$  \hfill (10)$$

where $y$ is the chemical, structural, or isotopic parameter, $A$ is a constant, $k$ is the rate constant, and $E_{dose}$ is the irradiated dose per volume. We normalized all integrated parameters, such as vibrational band intensities and isotopic abundances, to the starting value (non-irradiated tholin) when possible. The parameters for these fits are listed in Table 2. We note that the isotopic rate constant is closely matched by the infrared depletion and the amorphization (Raman broadening) rate constants, pointing at coupled structural and chemical effects of X-ray irradiation, which itself drives the isotopic fractionation, i.e., dehydrogenation of H over D bonds is accompanied by a change in the organic structure (creation of defects).

While thermal degradation (Bonnet et al. 2015) and X-ray irradiation of tholins both induce a similar loss of amines and alkyls (i.e., dehydrogenation of the tholin), there are important differences in their respective evolutions. Ramon analysis shows that thermal degradation leads to an increase of polyaromatic units (Figure 10), while X-irradiation does not affect the average size of polyaromatic units (even if infrared analysis shows that isolated aromatic units may be created). In both cases, defect formation competes with aromatization. For soft X-ray irradiation, the former is the dominant process.

Disks can be further irradiated by cosmic rays (protons and ions), although their effect may be modulated by winds and magnetic fields in T-Tauri systems (Cleeves et al. 2013). The effect of cosmic rays on organic matter has been studied via laboratory experiments, using low-energy or swift-heavy ion sources, whose energy can be deposited in solids through nuclear or electronic interactions, respectively. Brunetto et al. (2009) examined the effects of 200–400 keV ion irradiation on soots showing the destruction of large polyaromatic units. On the other hand, Costantini et al. (2005) reported the
while aromatization dominates over defect creation in thermally degraded heating. Sputtering stops as soon as the Si substrate is reached. Defect creation leads to amorphization in X-ray-irradiated tholins, while aromatization dominates over defect creation in thermally degraded tholins.

Figure 10. Raman parameters of irradiated tholins as a function of the X-ray dose at 0.5 keV compared to the evolution of thermally annealed (300–1000°C) tholins (Bonnet et al. 2015). (a) The size of polyaromatic units ($\times I_2/I_0$) is slightly affected by X-ray irradiation, while strongly affected by thermal heating. (b) Defect creation leads to amorphization in X-ray-irradiated tholins, while aromatization dominates over defect creation in thermally degraded tholins.

Figure 11. X-ray-irradiated (1.3 keV) tholin following SIMS analysis of irradiated regions (brighter spots indicated by arrows) and the reference (non-irradiated) regions. Sputtering stops as soon as the Si substrate is reached (blue spots).

carbonization of Kapton films and growth of polyaromatic structures induced by swift-heavy ion irradiation. Experiments by Sabri et al. (2015) similarly showed the graphitization of water-covered amorphous carbon induced by 200 keV protons. Together, these experiments point at the strong influence of the starting material structure on its evolution under ionizing radiation.

Recent disk models by Cleeves et al. (2016) including X-rays and cosmic ray ionization investigated the deuterium enrichment of organic molecules. They argued that since D/H ratios corresponding to the most D-enriched organic materials in the solar system were not attained in the disk, there is a need for some interstellar inheritance. However, these models did not include macromolecular organics. Our experiments show a net loss of heteroatoms from the organic matrix, implying that a fraction of organic solids could be transferred into volatiles (hydrocarbons, radicals, ions). We thus propose that radiolysis and fragmentation of organics can eventually transfer the high D/H ratios of enriched solids back into the gas phase. In addition, as seen by the differences from the irradiation tracks for two X-ray energies (0.5 and 1.3 keV) in Figure 12, extended studies with tholins of varying structure, using both broadband and hard X-ray sources, will be necessary to fully understand the effect of X-rays on organic matter.

6. Astrophysical Implications

Stellar X-ray flux density is an important ionization source in protoplanetary disks, impacting its photochemical evolution. In this experimental study, we have shown that soft X-rays can induce a large deuterium enrichment in protoplanetary refractory organics. We compared the non-thermal effects of X-rays to previous experiments of electron irradiation on polymers (Laurent et al. 2015). X-ray irradiation of tholins induces higher D-enrichment than electron irradiation of organic polymers. This may reflect a better efficiency of X-rays to induce H–D fractionation, or a tendency of tholins to accommodate more defects and hence record higher D-enrichment. The efficiency of irradiation will also depend on the size of the irradiated grains (Roskosz et al. 2016). Our experiments on $\mu$m thick tholins, comparable to the average size of observed dust grains in protoplanetary disks (from $\mu$m to cm sizes; Natta et al. 2007), demonstrate that photon-induced isotopic enrichment of organic solids in disks is possible.

As shown in Figure 15, protoplanetary disks undergo irradiation from T-Tauri stellar fields having strong X-ray and UV components, although the higher-energy X-ray photons are able to penetrate more effectively toward the disk midplane beyond 10 au (Nomura et al. 2007; Walsh et al. 2012). The X-ray spectra of the classical T-Tauri star, TW Hydrae, provides an integrated luminosity from 0.2 to 2 keV of $L_X = 2 \times 10^{30}$ erg s$^{-1}$ (Kastner et al. 2002). At 10 au, this amounts to a total exposure of $10^{36}$ eV cm$^{-2}$ in 1 Myr assuming an X-ray optical depth of $\tau_X = 1$. Using synchrotron soft X-rays, we have reached total fluences between $10^{22}$ and $10^{24}$ eV cm$^{-2}$, well within the lifetimes of protoplanetary disks. The experimental plateau for D-enrichment is achieved at fluences of $\sim 10^{24}$ eV cm$^{-2}$ providing an upper limit to D-enrichment of protoplanetary organic matter by soft X-rays.

Compared to UV photons and electron irradiation, soft X-rays and their secondary photoelectrons can strongly modify the bulk of $\mu$m-sized organic particles. In spite of its non-thermal nature, soft X-rays can affect the chemical, structural, and isotopic properties of organic macromolecules and solids via photochemical and radiolitical reactions, impacting the cycle of deuterium from dust to volatiles in cold regions of protoplanetary disks.
Figure 12. D/H and H/C abundances normalized to the initial tholin ($\alpha_{D/H}$ or $\alpha_{H/C}$) as a function of sputtering depth. For 1.3 keV the $\alpha_{H/C}$ profile after 10 minutes of irradiation was discarded due to large systematic errors assigned to localized hydrogenated inhomogeneities on the tholin surface (fractures/wedges), as seen in Figure 3.

Figure 13. Evolution of average normalized D/H vs. H/C ratios for X-ray-irradiated tholins at 0.5 and 1.3 keV.

Figure 14. Evolution of the normalized D/H abundances as a function of irradiation doses, for 30 keV electrons on organic polymers: polyethylene (PE), polystyrene (PS), and polyethylene terephthalate (PET; Laurent et al. 2015) and for 0.5 and 1.3 keV X-rays on tholins (this work).
X-ray experiments were performed at the SEXTANTS beamline of the SOLEIL synchrotron (project No. 20150772). We thank Ferenc Borondics for access to the IR and Raman spectromicroscopes of the SMIS beamline and François Nicolas for his help with diagnostics at the Surface Laboratory. N.C. and L.G. thank the European Research Council for funding via the ERC PrimChem project (grant agreement No. 636829).

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### Figure 15. Schema of the thermal and non-thermal processes on an XUV illuminated protoplanetary disk. UV photons and X-rays irradiate the disk (regions of ice formation and planetesimal growth) and outer edges. Ionization by cosmic rays (CRs) is represented in green.

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