Electron-Driven Proton Transfer Relieves Excited-State Antiaromaticity in Photoexcited DNA Base Pairs

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1. Computational Methods.

Equilibrium geometries for all base pairs, in the ground state (GS), locally-excited state (LE), and charge transfer state (CT), were optimized with Cs symmetry at (TD-)$\omega$B97XD/6-311+G(d,p) employing Gaussian16. Potential energy curves for the LE and CT states were constructed based on constrained geometry optimizations, in which selected N–H bonds were set to distances between 1Å and 2.5 Å, and varied at increments of 0.1 Å. Ground state potential energy curves were computed based on single point energies of the corresponding optimized CT state geometries. All energy points on the GS, LE, and CT curves were computed at CASPT2(8,8)/6-311+G(d,p) using an active space of eight orbitals (four occupied and four unoccupied π-orbitals, with eight electrons) employing Molpro2012.1. Electron density difference maps verified local excitations or charge-transfers for relevant structures (Figures S1 and S2).

Based on the equilibrium geometries of base pairs, dissected nucleus-independent chemical shifts, NICS(1)zz, were computed for each base, by splitting the optimized structures of base pairs into individual bases and considering each fragment separately (with the purported charge and electronic state). NICS(1)zz values were computed at 1 Å above the five and six membered ring centers and include only contributions from the "out-of-plane" (zz) tensor component perpendicular to the ring plane. All NICS(1)zz data were computed at CASSCF(10,10)/6-311+G(d,p) using an active space of ten orbitals (five occupied and five unoccupied π-orbitals, with ten electrons), employing the Dalton2016 program. Choice of active space for NICS(1)zz calculations was selected by comparing computed carbon shieldings at the CASSCF/6-311+G(d,p) level with different active space sizes (Tables S2, S3, S4 and S5). Computed NICS(1)zz at the PW91/IGLOIII level for separate bases in base pair structures were benchmarked against results for the corresponding base pairs and showed excellent agreement (Table S1). Multicenter indices (MCI) values were computed for base pairs structures at CASSCF(8,8)/6-311+G(d,p) using an active space of eight orbitals (four occupied π orbitals and four unoccupied π-orbitals, with eight electrons) employing ESI-3D.
2. Computed electron density difference (EDD) maps

Figure S1. Electron density ($\rho$) difference maps showing local excitations [$\rho$(LE) – $\rho$(GS)] (left) and charge transfer [$\rho$(CT) – $\rho$(GS)] (right) in the: A) Watson–Crick (WC) A•T base pair, B) pyrimidine excitation in the WC A•T base pair, and C) non-Watson–Crick (non-WC) A•T base pair. EDD maps were computed at CASSCF(8,8)/6-311+G(d,p) using equilibrium LE and CT state geometries.
Figure S2. Electron density ($\rho$) difference maps showing local excitations [$\rho$(LE) – $\rho$(GS)] (left) and charge transfer [$\rho$(CT) – $\rho$(GS)] (right) in the: A) Watson–Crick (WC) G•C base pair, B) pyrimidine excitation in the WC G•C base pair, C) non-Watson–Crick (non-WC) G•C base pair, and D) alternative proton transfer pathway in the WC G•C base pair. EDD maps were computed at CASSCF(8,8)/6-311+G(d,p) using equilibrium LE and CT state geometries.
3. **Computed NICS(1)\textsubscript{zz} values for base fragments in base pair structures vs. base pairs**

**Table S1.** Computed NICS(1)\textsubscript{zz} for base fragments in base pair structures vs. base pairs, in the ground state, at PW91/IGLOIII.

|                  | Purine (6MR) | Purine (5MR) | Pyrimidine |
|------------------|--------------|--------------|------------|
| WC A\textbullet T|              |              |            |
| base fragment\textsuperscript{a} | -20.4        | -26.7        | -2.2       |
| base pair        | -17.8        | -26.2        | -2.0       |
| non-WC A\textbullet T|            |              |            |
| base fragment\textsuperscript{a} | -20.4        | -26.7        | -2.2       |
| base pair        | -19.2        | -27.2        | -2.5       |
| WC G\textbullet C|              |              |            |
| base fragment\textsuperscript{a} | -5.9         | -24.0        | -3.9       |
| base pair        | -6.2         | -23.2        | -2.9       |
| non-WC G\textbullet C|          |              |            |
| base fragment\textsuperscript{a} | -5.9         | -24.1        | -3.8       |
| base pair        | -7.0         | -23.6        | -6.1       |

\textsuperscript{a}NICS(1)\textsubscript{zz} calculated by splitting the optimized structures of base pairs into two bases and considering each fragment separately.
4. Computed carbon shieldings at CASSCF/6-311+G(d,p), with different active spaces.

Based on ground state equilibrium structures of the WC A•T and G•C base pairs, carbon shieldings for the base fragments were computed to identify an appropriate active space for NICS calculations at the CASSCF level. Base fragments were considered so that a larger active space (up to 12,12) could be examined with the 6-311+G(d,p) basis set. Computed carbon shielding values converged at an active space size of 10 π-orbitals (five occupied and five unoccupied π-orbitals) and 10 electrons (10,10).

![Diagram of adenine (A), thymine (T), guanine (G), and cytosine (C).](image)

**Figure S3.** Carbon numbering for adenine (A), thymine (T), guanine (G), and cytosine (C).

**Table S2.** Computed carbon shieldings in the adenine (A) fragment (based on WC A•T geometry) at the CASSCF/6-311+G(d,p) level with different active space sizes.

| Atom / Shielding (ppm) | Active space size (π orbitals only) |
|------------------------|----------------------------------|
|                        | (2,2)  | (4,4)  | (6,6)  | (8,8)  | (10,10) | (12,12) |
| C1                     | 26.5   | 46.7   | 45.4   | 49.6   | 49.5    | 49.6    |
| C2                     | 27.1   | 36.7   | 35.6   | 42.7   | 44.5    | 44.4    |
| C3                     | 81.9   | 74.3   | 78.6   | 82.7   | 83.1    | 82.7    |
| C4                     | 30.9   | 40.5   | 37.6   | 48.4   | 48.4    | 50.8    |
| C5                     | 65.2   | 48.9   | 64.0   | 63.1   | 63.4    | 66.3    |

**Table S3.** Computed carbon shieldings in the thymine (T) fragment (based on WC A•T geometry) at the CASSCF/6-311+G(d,p) level with different active space sizes.

| Atom / Shielding (ppm) | Active space size (π orbitals only) |
|------------------------|----------------------------------|
|                        | (2,2)  | (4,4)  | (6,6)  | (8,8)  | (10,10) | (12,12) |
| C1                     | 19.2   | 35.6   | 35.7   | 35.7   | 37.1    | 37.1    |
| C2                     | 35.8   | 35.9   | 37.5   | 48.6   | 49.9    | 50.7    |
| C3                     | 92.5   | 93.9   | 93.5   | 94.8   | 94.5    | 93.8    |
| C4                     | 70.9   | 72.9   | 73.6   | 73.4   | 73.5    | 73.5    |
| C5                     | 181.2  | 181.3  | 181.3  | 181.3  | 181.3   | 181.1   |

**Table S4.** Computed carbon shieldings in the guanine (G) fragment (based on WC G•C geometry) at the CASSCF/6-311+G(d,p) level with different active space sizes.

| Atom / Shielding (ppm) | Active space size (π orbitals only) |
|------------------------|----------------------------------|
|                        | (2,2)  | (4,4)  | (6,6)  | (8,8)  | (10,10) | (12,12) |
| C1                     | 69.2   | 67.8   | 70.3   | 66.8   | 67.3    | 70.3    |
| C2                     | 82.6   | 85.9   | 85.0   | 83.0   | 83.4    | 83.6    |
| C3                     | 27.0   | 25.6   | 25.6   | 28.1   | 42.1    | 40.8    |
| C4                     | 30.6   | 31.9   | 31.9   | 46.8   | 46.6    | 45.5    |
Table S5. Computed carbon shieldings in the cytosine (C) fragment (based on WC G•C geometry) at the CASSCF/6-311+G(d,p) level with different active space sizes.

| Atom / Shielding (ppm) | Active space size (π orbitals only) |
|------------------------|-------------------------------------|
|                        | (2,2) | (4,4) | (6,6) | (8,8) | (10,10) |
| C₁                     | 59.9  | 63.3  | 64.8  | 65.1  | 65.2    |
| C₂                     | 110.2 | 109.8 | 109.6 | 110.9 | 111.2   |
| C₃                     | 16.6  | 29.1  | 30.7  | 32.3  | 34.5    |
| C₄                     | 30.9  | 30.5  | 33.0  | 44.7  | 44.8    |

C₅: 30.5  44.2  47.5  48.7  50.3  52.4
5. Computed $^1$H chemical shifts.

![Chemical Structures](image)

**Figure S4.** Hydrogen numbering for adenine (A), thymine (T), guanine (G), and cytosine (C).

**Table S6.** Computed $^1$H chemical shift in ppm for the Watson-Crick and non-Watson-Crick of A•T and G•C in different electronic states at the CASSCF(10,10)/6-311+G(d,p) level.

| Base Pair Structure | Ha | Hb | Hc |
|--------------------|----|----|----|
| A•T                | 7.3| 8.2| 6.0|
| A•T*               | 1.0| 3.0| 6.0|
| [A→T]*            | 8.7| 7.4| 4.4|
| G•C                | 6.9| 4.4| 6.4|
| G•C*               | 5.7| 4.8| 6.8|
| [G→C]*            | 7.3| 4.2| 4.4|
| A•T*               | 8.1| 7.3| 5.5|
| [A→T]*            | 3.7| 5.6| 6.2|
| G•C*               | 6.9| 4.6| 4.3|
| [G→C]*            | 6.5| 4.7| 6.3|
| non-WC A•T        | 8.1| 7.2| 5.9|
| non-WC A•T*       | 0.5| 2.6| 5.9|
| non-WC [A→T]*     | 7.8| 7.8| 3.4|
| non-WC G•C        | 6.9| 6.8| 6.4|
| non-WC G•C*       | 6.1| 4.6| 6.4|
| non-WC [G→C]*     | 7.3| 4.1| 4.1|

The base pair $^1$H chemical shifts were calculated using the following equation.

$$\delta_{BP} = \sigma_{benz} - \sigma_{BP} + \delta_{benz}$$

$\sigma_{BP}$ is the calculated shielding of the base pairs, $\sigma_{benz}$ is the calculated shielding for benzene, and $\delta_{benz}$ is the experimental H NMR shift (7.3 ppm) of benzene.
6. Computed GIMIC ring current plots

A)

Figure S5. Computed GIMIC ring current plots for the GS (left column), LE (middle column), and CT (right column) states of the: A) Watson–Crick A•T and G•C pairs (excitation on purines), B) Watson–Crick A•T and G•C pairs (excitation on the pyrimidines), and C) non-Watson–Crick A•T and G•C pairs (excitation on purines).
Table S7. HOMED values calculated at ωB97XD/6-311+G(d,p) for the Watson-Crick and non-Watson-Crick forms of A•T and G•C in different electronic states.

| Base Pair Structure | purine | pyrimidine |
|---------------------|--------|------------|
| A•T                 | 0.923  | 0.740      |
| A*•T                | 0.828  | 0.742      |
| [A→T]*              | 0.918  | 0.772      |
| G•C                 | 0.845  | 0.834      |
| G*•C                | 0.821  | 0.866      |
| [G→C]*              | 0.841  | 0.773      |
| A•T                 | 0.923  | 0.740      |
| A•T*                | 0.922  | 0.716      |
| [A⟵T]*            | 0.750  | 0.753      |
| G•C                 | 0.845  | 0.834      |
| G•C*                | 0.836  | 0.814      |
| [G⟵C]*            | 0.817  | 0.817      |
| non-WC A•T          | 0.940  | 0.732      |
| non-WC A*•T         | 0.820  | 0.736      |
| non-WC [A→T]*       | 0.904  | 0.511      |
| non-WC G•C          | 0.854  | 0.876      |
| non-WC G*•C         | 0.812  | 0.872      |
| non-WC [G→C]*       | 0.847  | 0.769      |

HOMED values were computed to quantify the effects of bond length alternation in the GS, LE, and CT states of the isolated base pairs. When excitation happens on the purines (A or G), the purine fragments show increased bond length alternation in the LE states (lower HOMED values compared to the GS states), and following PCET, the CT states regain bond length equalization character (HOMED values closer to the GS states). Note changes in the pyrimidine fragments; similar values for the GS and LE states, and increased bond length equalization in the CT state. When excitation happens on the pyrimidines (T or C), the purine fragments show minimal bond length alternation in the LE states (HOMED values close to the GS states), but much increased bond length alternation in the CT states (lower HOMED values compared to the GS states). Note also changes in the pyrimidine fragments; increased bond length alternation in the LE state, and bond length equalization in the CT state.
8. Multicenter indices (MCI) values

Multicenter indices (MCI) values were calculated for the $C_s$ geometries of base pairs, in the GS, LE, and CT states, at the CASSCF/6-311+G(d,p) level with an active space of eight $\pi$-orbitals (four occupied and four unoccupied $\pi$-orbitals, with eight electrons). MCI values were computed for the complete base pair structures (instead of base fragments), and show a loss of aromaticity upon excitation, with gain in aromaticity upon proton-coupled electron transfer. The MCI quantifies the extent of delocalized cyclic bonding—larger values reflect a more aromatic ring. For example, the MCI for benzene is 0.0435 at the CASSCF(6,6)/6-311++G(d,p) level.

Table S8. Computed MCI values at the CASSCF(8,8)/6-311+G(d,p) level for A•T.

| Base pair structure | Adenine (6MR) | Adenine (5MR) | Thymine |
|---------------------|---------------|---------------|---------|
| WC A•T (excitation on purine) | | | |
| A•T                 | 0.0196        | 0.0153        | 0.0018  |
| A•T*                | 0.0066        | 0.0129        | 0.0018  |
| [A→T]^*             | 0.0130        | 0.0193        | 0.0033  |
| WC A•T (excitation on pyrimidine) | | | |
| A•T                 | 0.0196        | 0.0153        | 0.0018  |
| A•T*                | 0.0177        | 0.0166        | 0.0021  |
| [A←T]^*             | 0.0073        | 0.0155        | 0.0019  |
| non-WC A•T (excitation on purine) | | | |
| non-WC A•T          | 0.0185        | 0.0194        | 0.0021  |
| non-WC A•T*         | 0.0064        | 0.0119        | 0.0022  |
| non-WC [A→T]^*      | 0.0088        | 0.0155        | 0.0054  |

Table S9. Computed MCI values at the CASSCF(8,8)/6-311+G(d,p) level for G•C.

| Base pair structure | Adenine (6MR) | Adenine (5MR) | Thymine |
|---------------------|---------------|---------------|---------|
| WC G•C (excitation on purine) | | | |
| G•C                 | 0.0054        | 0.0188        | 0.0058  |
| G•C*                | 0.0029        | 0.0091        | 0.0069  |
| [G→C]^*             | 0.0062        | 0.0131        | 0.0040  |
| WC G•C (excitation on pyrimidine) | | | |
| G•C                 | 0.0054        | 0.0188        | 0.0058  |
| G•C*                | 0.0060        | 0.0185        | 0.0046  |
| [G←C]^*             | 0.0055        | 0.0149        | 0.0050  |
| non-WC G•C (excitation on purine) | | | |
| non-WC G•C          | 0.0057        | 0.0191        | 0.0090  |
| non-WC G•C*         | 0.0031        | 0.0090        | 0.0088  |
| non-WC [G→C]^*      | 0.0063        | 0.0132        | 0.0096  |
| WC G•C (alternative PCET) | | | |
| G•C                 | 0.0054        | 0.0188        | 0.0058  |
| Alternative G•C*    | 0.0036        | 0.0119        | 0.0067  |
| Alternative [G→C]* | 0.0032 | 0.0139 | 0.0074 |
9. Computed MCI values for Watson–Crick A•T and G•C structures at $C_1$ symmetry

Since minima geometries of the LE and CT states of WC A•T and WC G•C are nonplanar, we also considered the effects of aromaticity gain and loss based on the $C_1$ structures. Minima geometries, energies, and MCI values of the GS, LE, and CT states were computed at the (TD-)ωB97XD/6-311+G(d,p) level and the trends agree with that observed for the planar models. Upon excitation to the LE state ($A^*•T$ and $G^*•C$), both base pairs show a loss of aromatic character in the six membered ring of the purine. Crossing to the CT state ([A→T]$^*$ and [G→C]$^*$) regains a large part of the aromatic character (Figure S4). Note the near planar geometry of the purines in base pairs at the CT state, indicating re-aromatization (Figure S5).

**Figure S6.** Energies and MCI values based on minima geometries of the GS, LE, and CT states, for: A) WC A•T, and B) WC G•C, at (TD-)ωB97XD/6-311+G(d,p).
Figure S7. Minima geometries for the LE and CT states of: A) WC A•T and B) WC G•C.
10. Alternative proton-coupled electron transfer pathway in the Watson–Crick G•C base pair.

We also considered an alternative PCET pathway in the WC G•C pair, involving excitation on the purine, and proton transfer from the exocyclic NH$_2$ group. Compared to transfer of the pyridinium proton, the resulting CT structure is higher in energy (3.0 eV relative to ground state G•C, compared to 2.6 eV for the pyridinium proton transfer), and computed NICS(1)$_{zz}$ values suggest less antiaromaticity relief (Figure S6). Consistent with our findings, studies based on dynamics simulations reported an expected longer excited-state lifetime for this proton transfer route. See: Francés-Monerris, A., Gattuso, H., Roca-Sanjuán, D., Tuñón, I., Marazzi, M., Dumont, E. & Monari, A. Dynamics of the excited-state hydrogen transfer in a (dG)(dC) homopolymer: intrinsic photostability of DNA. *Chem. Sci.* 9, 7902–7911 (2018).

![Potential energy functions](image)

*Figure S8.* Potential energy functions of the electronic ground state (GS), locally-excited $^1\pi\pi^*$ state (LE), and charge-transfer state (CT), with constrained N–H bond distance, for the Watson-Crick structure of G•C involving the exocyclic NH$_2$ of guanine at CASPT2(8,8)/6-311+G(d,p). NICS(1)$_{zz}$ values were computed for relevant structures at the CASSCF(10,10)/6-311+G(d,p) level.
11. Homolytic N–H $\sigma$-bond cleavage of adenine and guanine.

Computed energy profile for the homolytic N–H $\sigma$-bond cleavage of adenine and guanine at CASPT2(10,10)/6-311+G(d,p), in the ground state (GS) and $^1\pi\pi^*$ excited state (LE). Note that, for both adenine and guanine, the GS and LE curves for $\sigma$-bond cleavage parallel each other and are highly endothermic.

**Figure S9.** Potential energy functions for constrained N–H stretching (see N–H bond in bold) of the ground state (GS) and locally-excited $^1\pi\pi^*$ state (LE) of A) adenine and B) guanine at CASPT2(10,10)/6-311+G(d,p).
12. Computed vertical ionization potentials (IP) and electron affinities (EA)

Figure S10. Computed vertical ionization potentials (IP) in the $3\pi\pi^*$ states of pyrimidines (T and C) and vertical electron affinities (EA) in the ground states of purines (A and G), compared to their isomers (A', G', T', C') at \textit{ω}B97X-D/6-311+G(d,p). Dissected NICS(1)$_{zz}$ values were computed at PW91/IGLOIII.

We show in Figure S10 that the power to photo-oxidize a nucleobase (as evaluated by the triplet state IPs) and to reduce a nucleobase (as evaluated by the ground state EAs), can be related to, respectively, the excited-state antiaromatic character and the ground state aromatic character of the nucleobase considered. As shown in Figure S10 (top row), the higher triplet state IPs of thymine (T) ($3\pi\pi^*$ state IP: 6.2 eV) and cytosine (C) ($3\pi\pi^*$ state IP: 5.7 eV), compared to their fully cyclic [4$n$+2] $\pi$-electron delocalized isomers T' ($3\pi\pi^*$ state IP: 5.4 eV) and C' ($3\pi\pi^*$ state IP: 5.2 eV), can be rationalized by weaker excited-state antiaromaticity in the $3\pi\pi^*$ states of T and C (note more positive NICS(1)$_{zz}$ values in T' and C' than in T and C). Also (bottom row), the higher ground state EAs of thymine (A) (EA: 1.1 eV) and guanine (G) (EA: 0.7 eV) (i.e., difficulty to add an electron to the base), compared to their fully cyclic [4$n$+2] $\pi$-electron delocalized isomers A' (EA: 0.8 eV) and G' (EA: 0.5 eV), can be rationalized by greater aromaticity in the ground states of A and G (note more negative NICS(1)$_{zz}$ values for of A and G).
13. DNA duplex model study

Excited-state proton-coupled electron transfer in a G•C:C•G DNA duplex was considered, following the deactivation pathway proposed by Kohler and coworkers in *J. Am. Chem. Soc.* 137, 7059–7062 (2015) (see Figure S7). The studied G•C:C•G duplex structure was modified from a PDB file (1BNA) and computed using an ONIOM scheme. All four nucleobase structures were optimized at ωB97XD/6-31+G(d). The sugar phosphate backbone (see Figure S11, in grey) was optimized using a semiempirical method, PM6, with fixed atomic positions for all heavy atoms. Relative energies and multicenter index (MCI) analyses were computed at ωB97XD/6-31+G(d) for the resulting ONIOM optimized geometries.

![Optimized structure of guanine](image)

**Figure S11.** Proton-coupled electron transfer in the G•C:C•G DNA duplex.

Upon local excitation on guanine 1 (LE state), the energy of the CT and PCET states lower as an electron is transferred from guanine 1 to cytosine 3 through base stacking, and as a proton is transferred from guanine 4 to cytosine 3 (see relative energies in Table S10). In the LE state, guanine 1 (green color in LE state) loses aromaticity (note lower MCI value, Table S11) and planarity (Figure S12). Transferring an electron to cytosine 3 (red color in CT state) regains some aromatic character restores planarity (see Table S11 and Figure S11). Cytosine 3 loses a lesser degree of aromaticity) upon accepting an electron (cf. MCI value for 1 in the LE state.)

![Optimized structure of guanine](image)

**Figure S12.** Optimized structure of guanine 1 in the LE state and CT state. Note that 1 is non-planar in the LE state, but becomes planar upon electron transfer to cytosine 3.
Table S10. Relative energies (in eV) for the ground state, locally-excited (LE) state, charge transfer (CT) state, and proton-coupled electron transfer state (PCET) of the G•C:C•G duplex at (TD-)ωB97XD/6-31+G(d)//ONIOM(ωB97XD/6-31+G(d):PM6).

| State      | Relative energy (eV) |
|------------|----------------------|
| Ground state | 0                    |
| LE state    | 4.65                 |
| CT state    | 4.18                 |
| PCET state  | 3.93                 |

Table S11. Computed multicenter index (MCI) values for each ring system, at (TD-)ωB97XD/6-31+G(d)//ONIOM (ωB97XD/6-31+G(d):PM6).

| GS         | I_{6MR} | I_{5MR} | 2 | 3 | I_{6MR} | I_{5MR} |
|------------|---------|---------|---|---|---------|---------|
| LE state   | 0.0015  | 0.0153  | 0.0122 | 0.0116 | 0.0096  | 0.0288  |
| CT state   | 0.0052  | 0.0188  | 0.0123 | 0.0062 | 0.0093  | 0.0287  |
| PCET state | 0.0054  | 0.0193  | 0.0125 | 0.0044 | 0.0144  | 0.0277  |
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15. CASPT2(8,8)/6-311+G(d,p) energies for point on the potential energy curves.

**Table S12.** Computed total electronic energies (in Hartree) for the GS, LE, and CT states of WC A•T, as a function of N–H bond stretching (see bold-type N–H bond in Figure 2A).

| N–H bond (Å) | GS       | LE       | CT       |
|--------------|----------|----------|----------|
| Equilibrium  | -919.2237| —        | —        |
| 1.0          | —        | -919.0506| -919.0526|
| 1.1          | —        | -919.0397| -919.0662|
| 1.2          | —        | -919.0278| -919.0768|
| 1.3          | —        | -919.0496| -919.0860|
| 1.4          | —        | —        | -919.0934|
| 1.5          | —        | —        | -919.0988|
| 1.6          | —        | —        | -919.1025|
| 1.7          | —        | —        | -919.1049|
| 1.8          | —        | —        | -919.1062|
| 1.9          | -919.1415| —        | -919.1088|
| 2.0          | -919.1377| —        | -919.1045|
| 2.1          | -919.1333| —        | -919.1044|
| 2.2          | -919.1293| —        | -919.1041|
| 2.3          | -919.1219| —        | -919.1037|
| 2.4          | -919.1179| —        | -919.1032|
| 2.5          | -919.1137| —        | -919.1027|

**Table S13.** Computed total electronic energies (in Hartree) for the GS, LE, and CT states of WC A•T (excitation on pyrimidine), as a function of N–H bond stretching (see bold-type N–H bond in Figure 3A).

| N–H bond (Å) | GS       | LE       | CT       |
|--------------|----------|----------|----------|
| Equilibrium  | -919.2237| —        | —        |
| 1.0          | —        | -919.0593| —        |
| 1.1          | —        | -919.0594| —        |
| 1.2          | —        | -919.0541| —        |
| 1.3          | —        | -919.0500| —        |
| 1.4          | —        | —        | —        |
| 1.5          | —        | —        | —        |
| 1.6          | —        | —        | —        |
| 1.7          | —        | —        | -919.0264|
| 1.8          | —        | —        | -919.0298|
| 1.9          | —        | —        | -919.0319|
| 2.0          | —        | —        | -919.0330|
| 2.1          | -919.1649| —        | -919.0335|
| 2.2          | -919.1601| —        | -919.0336|
| 2.3          | -919.1544| —        | -919.0317|
| 2.4          | -919.1520| —        | -919.0309|
| 2.5          | -919.1494| —        | -919.0303|
Table S14. Computed total electronic energies (in Hartree) for the GS, LE, and CT states of non-WC A•T, as a function of N–H bond stretching (see bold-type N–H bond in Figure 4A).

| N–H bond (Å) | GS         | LE         | CT         |
|-------------|------------|------------|------------|
| Equilibrium | -919.2301  | —          | —          |
| 1.0         | —          | -919.0547  | —          |
| 1.1         | —          | -919.0530  | —          |
| 1.2         | —          | -919.0365  | -919.0390  |
| 1.3         | —          | —          | -919.0457  |
| 1.4         | —          | —          | -919.0512  |
| 1.5         | —          | —          | -919.0551  |
| 1.6         | —          | —          | -919.0575  |
| 1.7         | —          | —          | -919.0577  |
| 1.8         | -919.1627  | —          | -919.0591  |
| 1.9         | -919.1594  | —          | -919.0589  |
| 2.0         | -919.1560  | —          | -919.0571  |
| 2.1         | -919.1494  | —          | -919.0574  |
| 2.2         | -919.1462  | —          | -919.0565  |
| 2.3         | -919.1433  | —          | -919.0555  |
| 2.4         | -919.1404  | —          | -919.0545  |
| 2.5         | -919.1376  | —          | -919.0535  |

Table S15. Computed total electronic energies (in Hartree) for the GS, LE, and CT states of WC G•C, as a function of N–H bond stretching (see bold-type N–H bond in Figure 2B).

| N–H bond (Å) | GS         | LE         | CT         |
|-------------|------------|------------|------------|
| Equilibrium | -935.2596  | —          | —          |
| 1.0         | —          | -935.1007  | -935.1229  |
| 1.1         | —          | -935.1002  | -935.1281  |
| 1.2         | —          | —          | -935.1322  |
| 1.3         | —          | —          | -935.1383  |
| 1.4         | —          | —          | -935.1455  |
| 1.5         | —          | —          | -935.1520  |
| 1.6         | —          | —          | -935.1570  |
| 1.7         | —          | —          | -935.1605  |
| 1.8         | —          | —          | -935.1626  |
| 1.9         | -935.1869  | —          | -935.1634  |
| 2.0         | -935.1827  | —          | -935.1631  |
| 2.1         | -935.1780  | —          | -935.1621  |
| 2.2         | -935.1706  | —          | -935.1606  |
| 2.3         | -935.1652  | —          | -935.1587  |
| 2.4         | -935.1571  | —          | -935.1567  |
| 2.5         | -935.1481  | —          | -935.1546  |
Table S16. Computed total electronic energies (in Hartree) for the GS, LE, and CT states of WC G•C (excitation on pyrimidine), as a function of N–H bond stretching (see bold-type N–H bond in Figure 3B).

| N–H bond (Å) | GS       | LE       | CT       |
|--------------|----------|----------|----------|
| Equilibrium  | -935.2596| —        | —        |
| 1.0          | —        | -935.1088| —        |
| 1.1          | —        | -935.1054| —        |
| 1.2          | —        | -935.0987| —        |
| 1.3          | —        | -935.0875| —        |
| 1.4          | —        | —        | —        |
| 1.5          | —        | —        | —        |
| 1.6          | —        | —        | -935.0734|
| 1.7          | —        | —        | -935.0783|
| 1.8          | —        | —        | -935.0779|
| 1.9          | —        | —        | -935.0806|
| 2.0          | —        | —        | -935.0808|
| 2.1          | -935.1929| —        | -935.0789|
| 2.2          | -935.1892| —        | -935.0784|
| 2.3          | -935.1857| —        | -935.0778|
| 2.4          | -935.1820| —        | -935.0770|
| 2.5          | -935.1784| —        | -935.0759|

Table S17. Computed total electronic energies (in Hartree) for the GS, LE, and CT states of non-WC G•C, as a function of N–H bond stretching (see bold-type N–H bond in Figure 4B).

| N–H bond (Å) | GS       | LE       | CT       |
|--------------|----------|----------|----------|
| Equilibrium  | -935.2543| —        | —        |
| 1.0          | —        | -935.0986| —        |
| 1.1          | —        | -935.0976| —        |
| 1.2          | —        | -935.0909| -935.1071|
| 1.3          | —        | —        | -935.1129|
| 1.4          | —        | —        | -935.1180|
| 1.5          | —        | —        | -935.1217|
| 1.6          | —        | —        | -935.1238|
| 1.7          | —        | —        | -935.1248|
| 1.8          | -935.1842| —        | -935.1247|
| 1.9          | -935.1823| —        | -935.1241|
| 2.0          | -935.1803| —        | -935.1231|
| 2.1          | -935.1782| —        | -935.1219|
| 2.2          | -935.1762| —        | -935.1207|
| 2.3          | -935.1739| —        | -935.1195|
| 2.4          | -935.1718| —        | -935.1183|
| 2.5          | -935.1695| —        | -935.1171|
Table S18. Computed total electronic energies (in Hartree) for the GS, LE, and CT states of WC G•C (alternative PCET pathway), as a function of N–H bond stretching (see bold-type N–H bond in Figure S6).

| N–H bond (Å) | GS       | LE       | CT       |
|--------------|----------|----------|----------|
| Equilibrium  | -935.2596| —        | —        |
| 1.0          | —        | -935.1048| -935.1237|
| 1.1          | —        | -935.1026| -935.1316|
| 1.2          | —        | -935.0941| -935.1347|
| 1.3          | —        | —        | -935.1395|
| 1.4          | —        | —        | -935.1437|
| 1.5          | —        | —        | -935.1468|
| 1.6          | —        | —        | -935.1486|
| 1.7          | -935.1782| —        | -935.1494|
| 1.8          | -935.1752| —        | -935.1493|
| 1.9          | -935.1733| —        | -935.1487|
| 2.0          | -935.1690| —        | -935.1477|
| 2.1          | -935.1643| —        | -935.1465|
| 2.2          | -935.1615| —        | -935.1451|
| 2.3          | -935.1602| —        | -935.1436|
| 2.4          | -935.1558| —        | -935.1422|
| 2.5          | -935.1556| —        | -935.1408|
16. Optimized Cartesian coordinates for relevant structures.

**WC A•T \((C_6)\)**

|   | X         | Y         | Z         |
|---|-----------|-----------|-----------|
| C | -2.1184   | -0.73537  | 0.00000   |
| C | 1.83469   | 1.88203   | 0.00000   |
| N | -0.76941  | -0.70388  | 0.00000   |
| N | 0.45369   | 1.84824   | 0.00000   |
| C | -0.07423  | -1.85456  | 0.00000   |
| C | -0.39969  | 2.92550   | 0.00000   |
| C | -0.81199  | -3.05494  | 0.00000   |
| N | 0.25257   | 4.14640   | 0.00000   |
| N | -2.90148  | -1.79646  | 0.00000   |
| C | 2.44465   | 3.21174   | 0.00000   |
| C | -2.19362  | -2.93129  | 0.00000   |
| C | 1.62023   | 4.27387   | 0.00000   |
| C | -0.67035  | -3.45701  | 0.00000   |
| N | -2.17623  | -3.55764  | 0.00000   |
| C | -2.75917  | -4.02214  | 0.00000   |
| N | -4.11342  | -2.09158  | 0.00000   |
| C | -4.21717  | -0.71351  | 0.00000   |
| N | -3.00993  | -0.11339  | 0.00000   |
| C | -2.13367  | -1.11643  | 0.00000   |
| C | -0.68620  | -1.14717  | 0.00000   |
| N | -0.00742  | 0.00752   | 0.00000   |
| N | 2.79428   | -2.49744  | 0.00000   |
| C | 3.45629   | -1.28603  | 0.00000   |
| C | 4.91589   | -1.34887  | 0.00000   |
| C | 5.48521   | -2.56765  | 0.00000   |
| N | 4.74832   | -3.72544  | 0.00000   |
| C | 3.36346   | -3.75126  | 0.00000   |
| O | 2.82958   | -0.23143  | 0.00000   |
| C | 5.68526   | -0.6507   | 0.00000   |
| O | 2.73022   | -4.78165  | 0.00000   |
| H | -0.21784  | -4.38628  | 0.00000   |
| H | -4.86664  | -2.75759  | 0.00000   |
| H | -5.16456  | -0.19940  | 0.00000   |
| H | -0.53446  | 0.86772   | 0.00000   |
| H | 0.99256   | 0.01347   | 0.00000   |
| H | 1.75343   | -2.46845  | 0.00000   |
| H | 6.56040   | -2.70281  | 0.00000   |
| H | 5.19695   | -4.62701  | 0.00000   |
| H | 5.43538   | 0.53552   | 0.00000   |
| H | 6.76065   | -0.25386  | 0.00000   |
| H | 5.43538   | 0.53552   | -0.87798  |

\(N_{\text{im}} = 0\)

**WC A•T \((C_6)\)**

|   | X         | Y         | Z         |
|---|-----------|-----------|-----------|
| N | -0.06190  | -2.32303  | 0.00000   |
| C | -0.77035  | -3.45701  | 0.00000   |
| N | -2.17623  | -3.55764  | 0.00000   |
| C | -2.75917  | -4.02214  | 0.00000   |
| N | -4.11342  | -2.09158  | 0.00000   |
| C | -4.21717  | -0.71351  | 0.00000   |
| N | -3.00993  | -0.11339  | 0.00000   |
| C | -2.13367  | -1.11643  | 0.00000   |
| C | -0.68620  | -1.14717  | 0.00000   |
| N | -0.00742  | 0.00752   | 0.00000   |
| N | 2.79428   | -2.49744  | 0.00000   |
| C | 3.45629   | -1.28603  | 0.00000   |
| C | 4.91589   | -1.34887  | 0.00000   |
| C | 5.48521   | -2.56765  | 0.00000   |
| N | 4.74832   | -3.72544  | 0.00000   |
| C | 3.36346   | -3.75126  | 0.00000   |
| O | 2.82958   | -0.23143  | 0.00000   |
| C | 5.68526   | -0.6507   | 0.00000   |
| O | 2.73022   | -4.78165  | 0.00000   |
| H | -0.21784  | -4.38628  | 0.00000   |
| H | -4.86664  | -2.75759  | 0.00000   |
| H | -5.16456  | -0.19940  | 0.00000   |
| H | -0.53446  | 0.86772   | 0.00000   |
| H | 0.99256   | 0.01347   | 0.00000   |
| H | 1.75343   | -2.46845  | 0.00000   |
| H | 6.56040   | -2.70281  | 0.00000   |
| H | 5.19695   | -4.62701  | 0.00000   |
| H | 5.43538   | 0.53552   | 0.00000   |
| H | 6.76065   | -0.25386  | 0.00000   |
| H | 5.43538   | 0.53552   | -0.87798  |

\(N_{\text{im}} = 2\) (424\text{i} and 44\text{i} cm\text{\textsuperscript{-1}})

### WC [A → T]* (C₆)

|   | X    | Y     | Z     |
|---|------|-------|-------|
| C | -0.63276 | -3.40556 | 0.00000 |
| N | 0.03019  | -2.27092 | 0.00000 |
| C | -0.63545 | -1.07473 | 0.00000 |
| C | -2.05635 | -1.13800 | 0.00000 |
| N | -1.97678 | -3.57056 | 0.00000 |
| C | -2.62290 | -2.42503 | 0.00000 |
| N | -3.97958 | -2.20115 | 0.00000 |
| C | -2.05635 | -1.13800 | 0.00000 |
| N | -1.97678 | -3.57056 | 0.00000 |
| C | -2.62290 | -2.42503 | 0.00000 |
| N | -3.97958 | -2.20115 | 0.00000 |
| C | -4.14994 | -0.84696 | 0.00000 |
| N | 0.09834  | 0.01953  | 0.00000 |
| H | -5.12903 | -0.39784 | 0.00000 |
| H | -0.03278 | -4.30660 | 0.00000 |
| C | -3.51151 | -1.13195 | 0.00000 |
| N | 2.84270  | 0.03031  | 0.00000 |
| C | -0.67977 | 1.24214  | 0.00000 |
| O | 0.98142  | -0.01887 | 0.00000 |
| H | -0.01856 | -0.01263 | 0.00000 |
| C | -0.65075 | -1.20972 | 0.00000 |
| O | -0.09382 | -2.29387 | 0.00000 |
| H | -2.47289 | -2.09021 | 0.00000 |
| H | -2.57116 | 3.08877  | 0.87074 |
| H | -2.57116 | 3.08877  | -0.87074 |
| H | -3.93271 | 2.32709  | 0.00000 |

### WC A+T* (C₆)

|   | X    | Y     | Z     |
|---|------|-------|-------|
| C | 3.51151 | -1.13195 | 0.00000 |
| N | 2.84270  | 0.03031  | 0.00000 |
| C | 3.52701  | 1.18979  | 0.00000 |
| C | 4.93528  | 1.10951  | 0.00000 |
| N | 4.82081  | -1.32622 | 0.00000 |
| C | 5.48359  | -0.16414 | 0.00000 |
| N | 6.83873  | 0.04212  | 0.00000 |
| N | 5.91745  | 2.07931  | 0.00000 |
| C | 7.03022  | 1.40298  | 0.00000 |
| N | 2.86467  | 2.34931  | 0.00000 |
| H | 8.02081  | 1.83376  | 0.00000 |
| H | 2.88541  | -2.01991 | 0.00000 |
| H | 3.39143  | 3.20546  | 0.00000 |
| H | 1.83967  | 2.36870  | 0.00000 |
| H | 7.54427  | -0.67528 | 0.00000 |
| N | -2.04999 | -1.17184 | 0.00000 |
| C | -2.81554 | -0.04082 | 0.00000 |
| H | -3.88951 | -0.14132 | 0.00000 |
| C | -2.10710 | 1.20921  | 0.00000 |
| C | -2.85280 | 2.48320  | 0.00000 |
| C | -0.67977 | 1.24214  | 0.00000 |
| O | 0.00593  | 2.28422  | 0.00000 |
| N | -0.01856 | -0.01263 | 0.00000 |
| H | 0.98142  | -0.01887 | 0.00000 |
| C | -0.65075 | -1.20972 | 0.00000 |
| O | -0.09382 | -2.29387 | 0.00000 |
| H | -2.47289 | -2.09021 | 0.00000 |
| H | -2.57116 | 3.08877  | 0.87074 |
| H | -2.57116 | 3.08877  | -0.87074 |
| H | -3.93271 | 2.32709  | 0.00000 |

$N_{im} = 1 \ (26 \mathrm{ \text{cm}^{-1}})$

$N_{im} = 4 \ (301\mathrm{i}, 187\mathrm{i}, 118\mathrm{i}, \text{and } 66 \mathrm{ \text{cm}^{-1}})$
| X    | Y    | Z    |
|------|------|------|
| C    | 3.60287 | -1.45809 | 0.00000 |
| N    | 3.09697 | -0.12368 | 0.00000 |
| C    | 3.87895 | 0.97162 | 0.00000 |
| C    | 5.23758 | 0.73859 | 0.00000 |
| N    | 5.67196 | -0.62086 | 0.00000 |
| N    | 7.05637 | -0.50213 | 0.00000 |
| N    | 6.31486 | 1.61995 | 0.00000 |
| C    | 7.35373 | 0.85019 | 0.00000 |
| N    | 3.32158 | 2.19957 | 0.00000 |
| H    | 8.37772 | 1.19524 | 0.00000 |
| H    | 2.88700 | -2.25349 | 0.00000 |
| H    | 3.93708 | 2.99262 | 0.00000 |
| H    | 7.70761 | -1.26624 | 0.00000 |
| N    | -2.16081 | -1.00127 | 0.00000 |
| C    | -2.76250 | 0.22199 | 0.00000 |
| H    | -3.84546 | 0.21640 | 0.00000 |
| C    | -2.04470 | 1.36903 | 0.00000 |
| C    | -2.65924 | 2.72932 | 0.00000 |
| O    | -0.58333 | 1.26553 | 0.00000 |
| N    | -0.49360 | 1.29033 | 0.00000 |
| C    | -0.00361 | 0.00225 | 0.00000 |
| N    | -1.96727 | -1.39044 | 0.00000 |
| H    | -2.36795 | -2.31184 | 0.00000 |
| C    | -2.55563 | -0.57587 | 0.00000 |
| C    | 0.13426 | -2.35758 | 0.00000 |
| C    | 1.45897 | -2.21475 | 0.00000 |
| H    | 2.02795 | -3.13988 | 0.00000 |
| N    | 2.16829 | -1.08997 | 0.00000 |
| C    | 1.38444 | 0.00075 | 0.00000 |
| H    | 2.71890 | 1.64908 | 0.00000 |
| N    | 4.99463 | -0.50955 | 0.00000 |
| C    | 5.93306 | -1.51310 | 0.00000 |
| H    | 5.51889 | -2.51465 | 0.00000 |
| C    | 7.25989 | -1.29086 | 0.00000 |
| C    | 8.29272 | -2.37385 | 0.00000 |
| C    | 7.71927 | 0.09711 | 0.00000 |
| O    | 8.87679 | 0.45564 | 0.00000 |
| N    | 6.68723 | 1.04610 | 0.00000 |
| H    | 6.96861 | 2.01688 | 0.00000 |
| C    | 5.33445 | 0.81096 | 0.00000 |
| O    | 4.51998 | 1.72669 | 0.00000 |
| H    | 3.98693 | -0.74935 | 0.00000 |
| H    | 7.82491 | -3.36022 | 0.00000 |
| H    | 8.93837 | -2.29160 | -0.87777 |
| H    | 8.93837 | 2.29160 | 0.87777 |

\( N_m = 2 \) (425i and 75i cm\(^{-1}\))

| X    | Y    | Z    |
|------|------|------|
| N    | 1.74918 | 1.31393 | 0.00000 |
| C    | 0.58241 | 2.03122 | 0.00000 |
| H    | 0.58871 | 3.11124 | 0.00000 |
| N    | -0.49360 | 1.29033 | 0.00000 |
| C    | -0.00361 | 0.00225 | 0.00000 |
| C    | -0.62922 | -1.25565 | 0.00000 |
| N    | -1.96727 | -1.39044 | 0.00000 |
| H    | -2.36795 | -2.31184 | 0.00000 |
| C    | -2.55563 | -0.57587 | 0.00000 |
| N    | 0.13426 | -2.35758 | 0.00000 |
| C    | 1.45897 | -2.21475 | 0.00000 |
| H    | 2.02795 | -3.13988 | 0.00000 |
| N    | 2.16829 | -1.08997 | 0.00000 |
| C    | 1.38444 | 0.00075 | 0.00000 |
| H    | 2.71890 | 1.64908 | 0.00000 |
| N    | 4.99463 | -0.50955 | 0.00000 |
| C    | 5.93306 | -1.51310 | 0.00000 |
| H    | 5.51889 | -2.51465 | 0.00000 |
| C    | 7.25989 | -1.29086 | 0.00000 |
| C    | 8.29272 | -2.37385 | 0.00000 |
| C    | 7.71927 | 0.09711 | 0.00000 |
| O    | 8.87679 | 0.45564 | 0.00000 |
| N    | 6.68723 | 1.04610 | 0.00000 |
| H    | 6.96861 | 2.01688 | 0.00000 |
| C    | 5.33445 | 0.81096 | 0.00000 |
| O    | 4.51998 | 1.72669 | 0.00000 |
| H    | 3.98693 | -0.74935 | 0.00000 |
| H    | 7.82491 | -3.36022 | 0.00000 |
| H    | 8.93837 | -2.29160 | -0.87777 |
| H    | 8.93837 | 2.29160 | 0.87777 |

\( N_m = 0 \)

| X    | Y    | Z    |
|------|------|------|
| N    | 29  |      |      |

non-WC A*T (C\(_s\))
\[ \begin{array}{ccc}
\text{X} & \text{Y} & \text{Z} \\
\hline
\text{N} & -0.00692 & 0.01049 & 0.00000 \\
\text{C} & -0.87506 & 1.06027 & 0.00000 \\
\text{H} & -0.54924 & 2.08789 & 0.00000 \\
\text{N} & -2.16867 & 0.66138 & 0.00000 \\
\text{C} & -2.10874 & -0.66846 & 0.00000 \\
\text{C} & -3.13116 & -1.70439 & 0.00000 \\
\text{N} & -4.42907 & -1.33811 & 0.00000 \\
\text{H} & -5.14907 & -2.04244 & 0.00000 \\
\text{H} & -4.67059 & -0.35969 & 0.00000 \\
\text{N} & -2.16867 & -2.97668 & 0.00000 \\
\text{C} & -1.48053 & -3.29548 & 0.00000 \\
\text{H} & -1.21864 & -4.34441 & 0.00000 \\
\text{C} & -0.37833 & -2.38980 & 0.00000 \\
\text{C} & -0.76639 & -1.14261 & 0.00000 \\
\text{N} & 2.45236 & -2.78004 & 0.00000 \\
\text{C} & 3.01321 & -4.03327 & 0.00000 \\
\text{H} & 2.29539 & -4.84542 & 0.00000 \\
\text{C} & 4.34082 & -4.25503 & 0.00000 \\
\text{C} & 4.96452 & -5.61546 & 0.00000 \\
\text{H} & 4.20064 & -6.39543 & 0.00000 \\
\text{C} & 5.22643 & -3.09264 & 0.00000 \\
\text{O} & 6.43811 & -3.12948 & 0.00000 \\
\text{N} & 4.55941 & -1.85953 & 0.00000 \\
\text{H} & 5.14097 & -1.03312 & 0.00000 \\
\text{H} & 1.41834 & -2.67752 & 0.00000 \\
\text{H} & 0.99304 & 0.01982 & 0.00000 \\
\text{H} & 5.6175 & -0.74841 & -0.87779 \\
\text{H} & 5.6175 & -0.74841 & 0.87779 \\
\end{array} \]

\[ N_m = 2 \text{ (273i and 34i cm}^{-1}\text{) } \]

\[ \begin{array}{ccc}
\text{X} & \text{Y} & \text{Z} \\
\hline
\text{C} & -1.43127 & -3.27196 & 0.00000 \\
\text{C} & 5.50941 & -2.40423 & 0.00000 \\
\text{N} & -2.73215 & -3.02646 & 0.00000 \\
\text{N} & 4.69551 & -1.16348 & 0.00000 \\
\text{C} & -3.13558 & -1.74598 & 0.00000 \\
\text{C} & 3.36474 & -1.14701 & 0.00000 \\
\text{C} & -2.14849 & -0.71064 & 0.00000 \\
\text{N} & 2.70079 & -2.28052 & 0.00000 \\
\text{N} & -0.39725 & -2.3094 & 0.00000 \\
\text{C} & 4.75448 & -3.56747 & 0.00000 \\
\text{C} & -0.78019 & -1.13861 & 0.00000 \\
\text{C} & 3.34390 & -3.53431 & 0.00000 \\
\text{N} & 0.00290 & -0.00671 & 0.00000 \\
\text{N} & -2.22308 & 0.60140 & 0.00000 \\
\text{C} & -0.87832 & 0.96599 & 0.00000 \\
\text{N} & -4.43168 & -1.47123 & 0.00000 \\
\text{H} & -0.59774 & 2.00928 & 0.00000 \\
\text{H} & -1.14977 & -4.31790 & 0.00000 \\
\text{O} & 6.72678 & -2.22487 & 0.00000 \\
\text{H} & -4.74716 & -0.51397 & 0.00000 \\
\text{H} & 5.20147 & -0.29356 & 0.00000 \\
\text{O} & 2.78179 & 0.03935 & 0.00000 \\
\text{H} & 2.69813 & -4.39075 & 0.00000 \\
\text{H} & 1.69305 & -2.27201 & 0.00000 \\
\text{C} & 5.47844 & -4.89020 & 0.00000 \\
\text{H} & 1.80290 & -0.00698 & 0.00000 \\
\text{H} & -5.09669 & -2.22862 & 0.00000 \\
\text{H} & 4.77087 & -5.72302 & 0.00000 \\
\text{H} & 6.12274 & -4.99076 & -0.87846 \\
\text{H} & 6.12274 & -4.99076 & 0.87846 \\
\end{array} \]

\[ N_m = 4 \text{ (225i, 71i, 35i, and 10i cm}^{-1}\text{) } \]

\[ WC G\cdot C (C_s) \]

\[ WC G\cdot C (C_s) \]
| X   | Y       | Z   |
|-----|---------|-----|
| N   | 4.31228 | 0.91516 | 0.00000 |
| C   | 4.92093 | -0.29747 | 0.00000 |
| H   | 6.00363 | -0.29639 | 0.00000 |
| C   | 4.18619 | -1.42735 | 0.00000 |
| H   | 4.65195 | -2.40205 | 0.00000 |
| C   | 2.75261 | -1.27382 | 0.00000 |
| H   | 4.65195 | -2.40205 | 0.00000 |
| N   | 4.18619 | -1.42735 | 0.00000 |
| C   | 2.75261 | -1.27382 | 0.00000 |

N\textsubscript{m} = 0

**WC G\textsuperscript{*}C (C\textsubscript{4})**

| X   | Y       | Z   |
|-----|---------|-----|
| N   | -4.01388 | -0.07995 | 0.00000 |
| C   | -4.17598 | -1.42810 | 0.00000 |
| H   | -5.15594 | -1.88256 | 0.00000 |
| N   | -3.03905 | -2.10906 | 0.00000 |
| C   | -2.06322 | -1.16705 | 0.00000 |
| H   | -0.63898 | -1.28498 | 0.00000 |
| O   | 0.05831  | -2.30304 | 0.00000 |
| N   | -0.00084 | 0.00283  | 0.00000 |
| H   | 0.99916  | -0.00298 | 0.00000 |
| C   | -0.67208 | 1.18789  | 0.00000 |
| N   | 0.05418  | 2.30209  | 0.00000 |
| H   | 1.07416  | 2.29505  | 0.00000 |
| H   | -0.44984 | 3.17292  | 0.00000 |
| N   | -2.01702 | 1.30936  | 0.00000 |
| C   | -2.65236 | 0.15287  | 0.00000 |
| H   | -4.73677 | 0.61979  | 0.00000 |
| N   | 4.83915  | 1.29964  | 0.00000 |
| C   | 5.63688  | 0.18388  | 0.00000 |
| H   | 6.70549  | 0.34666  | 0.00000 |
| C   | 5.06048  | -1.04570 | 0.00000 |
| H   | 5.66524  | -1.94173 | 0.00000 |
| C   | 3.63865  | -1.11010 | 0.00000 |
| N   | 2.99245  | -2.29205 | 0.00000 |
| H   | 1.97198  | -2.32411 | 0.00000 |
| H   | 3.50839  | -3.15219 | 0.00000 |
| N   | 2.87896  | 0.00266  | 0.00000 |
| C   | 3.44851  | 1.21786  | 0.00000 |
| O   | 2.80445  | 2.27756  | 0.00000 |
| H   | 5.22750  | 2.22801  | 0.00000 |

N\textsubscript{m} = 4 (406i, 162i, 68i, and 5i cm\textsuperscript{-1})

**WC [G\textsuperscript{*}C] (C\textsubscript{4})**

| X   | Y       | Z   |
|-----|---------|-----|
| N   | -3.99262 | 0.23594 | 0.00000 |
N_{Im} = 1 (248 cm^{-1})

|   | X    | Y    | Z    |
|---|------|------|------|
| C | -4.09120 | 1.59226 | 0.00000 |
| H | -5.04717 | 2.09388 | 0.00000 |
| N | -2.92402 | 2.21085 | 0.000000 |
| C | -2.01175 | 1.22048 | 0.00000 |
| C | -0.54810 | 1.23692 | 0.00000 |
| O | 0.09946 | 2.27393 | 0.00000 |
| N | 0.04101 | -0.00224 | 0.00000 |
| H | 1.94099 | -0.01001 | 0.00000 |
| C | -0.68068 | -1.11046 | 0.00000 |
| N | -0.06184 | -2.28826 | 0.00000 |
| H | 0.95663 | -2.34422 | 0.00000 |
| C | 5.69470 | -0.35440 | 0.00000 |
| N | 6.74832 | -0.56741 | 0.00000 |
| C | 5.13133 | 0.91111 | 0.00000 |
| H | 5.76922 | 1.78357 | 0.00000 |
| C | 3.76411 | 1.06583 | 0.00000 |
| N | 3.08999 | 2.25228 | 0.00000 |
| H | 2.08374 | 2.29662 | 0.00000 |
| C | 3.60938 | 3.10792 | 0.00000 |
| N | 2.95944 | -0.08537 | 0.00000 |
| C | 3.46664 | -1.35441 | 0.00000 |
| O | 2.74193 | -2.35145 | 0.00000 |
| H | 5.17107 | -2.39944 | 0.00000 |

N_{Im} = 4 (405i, 112i, 75i, and 61i cm^{-1})

|   | X    | Y    | Z    |
|---|------|------|------|
| C | -3.76414 | -2.29388 | 0.00000 |
| C | -3.96935 | -3.63896 | 0.00000 |
| H | -4.96083 | -4.06582 | 0.00000 |
|     | X     | Y     | Z     |
|-----|-------|-------|-------|
| N   | 6.98515 | -1.94172 | 0.00000 |
| C   | 7.10630 | -0.56728 | 0.00000 |
| H   | 8.07272 | -0.08522 | 0.00000 |
| N   | 5.95897 | 0.04074  | 0.00000 |
| C   | 5.02818 | -0.97648 | 0.00000 |
| C   | 3.91964 | -3.33808 | 0.00000 |
| N   | 3.24110 | -4.52780 | 0.00000 |
| H   | 2.23745 | -4.56727 | 0.00000 |
| H   | 3.75010 | -5.38994 | 0.00000 |
| N   | 3.10021 | -2.19136 | 0.00000 |
| C   | 3.71171 | -1.04483 | 0.00000 |
| O   | 3.06214 | 0.10345  | 0.00000 |
| H   | 5.09972 | 0.05580  | 0.00000 |
| H   | 1.24292 | -2.34011 | 0.00000 |
| H   | 2.07868 | -0.02526 | 0.00000 |

N_{im}= 1 (281 \text{ cm}^{-1})

| X     | Y     | Z     |
|-----|-------|-------|
| N   | 6.95821 | 2.16758 | 0.00000 |
| C   | 7.18283 | 0.81848 | 0.00000 |
| H   | 8.17877 | 0.40415 | 0.00000 |
| N   | 6.06843 | 0.12333 | 0.00000 |
| C   | 5.06441 | 1.06669 | 0.00000 |

N_{im}= 4 (168i, 80i, 68i, and 11i \text{ cm}^{-1})
|     | X      | Y      | Z     |
|-----|--------|--------|-------|
| C   | -0.40965 | -1.29519 | 0.00000 |
| N   | 0.01878  | 0.03411  | 0.00000 |
| C   | -0.78700  | 1.13362  | 0.00000 |
| N   | -2.09820  | 1.08159  | 0.00000 |
| C   | -2.55249  | -0.18860  | 0.00000 |
| C   | -1.83397  | -1.37777  | 0.00000 |
| N   | -2.68136  | -2.46641  | 0.00000 |
| C   | -3.87270  | -1.95159  | 0.00000 |
| N   | -3.86148  | -0.57155  | 0.00000 |
| N   | -0.14853  | 2.32579  | 0.00000 |
| O   | 0.42553  | -2.19986  | 0.00000 |
| H   | 1.04406  | 0.18903  | 0.00000 |
| H   | 0.85720  | 2.37711  | 0.00000 |
| H   | -0.70214  | 3.16244  | 0.00000 |
| H   | -4.65268  | 0.04998  | 0.00000 |
| H   | -4.79792  | -2.50866  | 0.00000 |
| N   | 3.15462  | -1.62491  | 0.00000 |
| C   | 3.55669  | -0.28055  | 0.00000 |
| N   | 4.88239  | -0.01833  | 0.00000 |
| C   | 5.75120  | -1.01126  | 0.00000 |
| C   | 5.37379  | -2.39058  | 0.00000 |
| C   | 4.04191  | -2.63833  | 0.00000 |
| O   | 2.68944  | 0.59098  | 0.00000 |
| N   | 7.06006  | -0.67485  | 0.00000 |
| H   | 2.13939  | -1.83275  | 0.00000 |
| H   | 3.62222  | -3.63677  | 0.00000 |
| H   | 6.10045  | -3.19004  | 0.00000 |
| H   | 7.78564  | -1.36718  | 0.00000 |
| H   | 7.30003  | 0.30175  | 0.00000 |

Non-WC G•C (C−)

|     | X      | Y      | Z     |
|-----|--------|--------|-------|
| C   | -0.68422  | -1.27631  | 0.00000 |
| N   | -0.00429  | -0.00706  | 0.00000 |
| C   | -0.65254  | 1.19539  | 0.00000 |
| N   | -1.98037  | 1.36349  | 0.00000 |
| C   | -2.65509  | 0.22118  | 0.00000 |
| C   | -2.09259  | -1.12233  | 0.00000 |
| N   | -3.10045  | -2.03538  | 0.00000 |
| C   | -4.21882  | -1.32253  | 0.00000 |

N_{im}= 1 (71/cm²)
\[
\begin{array}{ccc}
N & -4.02620 & 0.01997 \\
N & 0.14050 & 2.28266 \\
O & 0.02546 & -2.30283 \\
H & 0.99571 & -0.00938 \\
H & 1.14532 & 2.18408 \\
H & -0.28423 & 3.19323 \\
H & -4.73362 & 0.73456 \\
N & -5.20975 & -1.75326 \\
N & 2.79223 & -2.08609 \\
C & 3.42522 & -0.84034 \\
N & 4.77940 & -0.81021 \\
C & 5.45855 & -1.94108 \\
C & -0.71479 & -1.17982 \\
N & 0.00002 & 0.00006 \\
C & -0.59567 & 1.17202 \\
N & -1.96131 & 1.42931 \\
C & -2.65461 & 0.33291 \\
C & -2.16404 & -1.00660 \\
N & -3.17725 & -1.89083 \\
C & -4.27019 & -1.14887 \\
N & -4.02539 & 0.19013 \\
N & 0.15698 & 2.28136 \\
O & -0.16189 & -2.26777 \\
H & 1.16210 & 2.21052 \\
H & -0.29210 & 3.17963 \\
H & -4.70092 & 0.93654 \\
H & -5.27485 & -1.54384 \\
N & 2.95857 & -2.01456 \\
C & 3.55314 & -0.79532 \\
N & 4.82074 & -0.58425 \\
C & 5.63923 & -1.71527 \\
C & 5.13515 & -2.99040 \\
C & 3.74996 & -3.18536 \\
O & 2.73957 & 0.26523 \\
N & 6.98005 & -1.40330 \\
H & 1.95741 & -2.10221 \\
H & 3.23731 & -4.12787 \\
H & 5.79499 & -3.84810 \\
H & 7.68585 & -2.11236 \\
H & 7.24010 & -0.43619 \\
H & 1.80002 & 0.00007 \\
\end{array}
\]

\(N_{\text{Im}} = 4\) (443, 340, 92, and 671 cm\(^{-1}\))

\[
\begin{array}{ccc}
X & Y & Z \\
\text{non-WC [G→C]* (C\textsubscript{1})} \\
C & -0.71479 & -1.17982 \\
N & 0.00002 & 0.00006 \\
C & -0.59567 & 1.17202 \\
N & -1.96131 & 1.42931 \\
C & -2.65461 & 0.33291 \\
C & -2.16404 & -1.00660 \\
N & -3.17725 & -1.89083 \\
C & -4.27019 & -1.14887 \\
N & -4.02539 & 0.19013 \\
N & 0.15698 & 2.28136 \\
O & -0.16189 & -2.26777 \\
H & 1.16210 & 2.21052 \\
H & -0.29210 & 3.17963 \\
H & -4.70092 & 0.93654 \\
H & -5.27485 & -1.54384 \\
N & 2.95857 & -2.01456 \\
C & 3.55314 & -0.79532 \\
N & 4.82074 & -0.58425 \\
C & 5.63923 & -1.71527 \\
C & 5.13515 & -2.99040 \\
C & 3.74996 & -3.18536 \\
O & 2.73957 & 0.26523 \\
N & 6.98005 & -1.40330 \\
H & 1.95741 & -2.10221 \\
H & 3.23731 & -4.12787 \\
H & 5.79499 & -3.84810 \\
H & 7.68585 & -2.11236 \\
H & 7.24010 & -0.43619 \\
H & 1.80002 & 0.00007 \\
\end{array}
\]

\(N_{\text{Im}} = 1\) (482 and 13 cm\(^{-1}\))

\[
\begin{array}{ccc}
X & Y & Z \\
\text{WC A*•T (C\textsubscript{1})} \\
O & -1.85163 & 2.13679 \\
C & -2.64260 & 1.19814 \\
N & -2.19055 & -0.10195 \\
C & -2.96198 & -1.24032 \\
O & -2.51328 & -2.36072 \\
N & -4.32373 & -0.98644 \\
C & -4.85729 & 0.27461 \\
C & -4.09209 & 1.37955 \\
N & 0.97289 & 1.88488 \\
C & 1.44222 & 0.64314 \\
\end{array}
\]
| Element | X   | Y   | Z   |
|---------|-----|-----|-----|
| N       | 0.61342 | -0.39453 | 0.06856 |
| C       | 1.19328 | -1.62385 | 0.28749 |
| N       | 2.46947 | -1.84463 | -0.25278 |
| C       | 3.30695 | -0.87296 | -0.02351 |
| H       | 5.92561 | 1.01594 | -0.27709 |
| H       | 0.55181 | -2.47786 | 0.45161 |
| H       | 1.62187 | 2.64857 | 0.24920 |
| H       | -0.03384 | 2.05337 | 0.03516 |
| H       | -1.15265 | -0.24538 | 0.00333 |
| H       | -5.93974 | 0.31960 | -0.12288 |
| H       | 5.27668 | -1.43182 | -0.61525 |
| H       | -4.19174 | -1.79999 | 0.02954 |
| H       | -4.63485 | 2.77135 | -0.23442 |
| H       | -4.30620 | 3.37066 | 0.61891 |
| H       | -4.26721 | 3.27169 | -1.1361 |
| H       | -5.72646 | 2.76435 | -0.25958 |

Sum of electronic and thermal free energy: –921.03406 a.u.

WC [A→T]* (C₁)

| X     | Y     | Z     |
|-------|-------|-------|
| C     | -1.92895 | -1.88714 | 0.75402 |
| C     | -2.56210 | -0.81095 | 0.24595 |
| N     | -1.92757 | 0.43296 | 0.36486 |
| C     | -2.49498 | 1.62638 | -0.01132 |
| O     | -1.90136 | 2.68879 | 0.06034 |
| N     | -3.78104 | 1.51806 | -0.46845 |
| C     | -4.48967 | 0.30759 | -0.55488 |
| C     | -3.87602 | -0.86732 | -0.18811 |
| N     | 0.92320 | -1.72515 | 0.29857 |
| C     | 1.65856 | -0.63002 | 0.30613 |
| N     | 1.04141 | 0.54246 | 0.65009 |
| C     | 1.71467 | 1.67104 | 0.63839 |
| N     | 3.02148 | 1.85099 | 0.33180 |
| C     | 3.62099 | 0.72561 | 0.01341 |
| C     | 3.03715 | -0.55258 | -0.03146 |
| N     | 3.94608 | -1.49651 | -0.40953 |
| C     | 5.05475 | -0.82091 | -0.59641 |
| N     | 4.92994 | 0.51733 | -0.35312 |
| H     | 5.99428 | -1.25578 | -0.89777 |
| H     | 1.15974 | 2.56594 | 0.89387 |
| H     | 1.50715 | -2.51651 | 0.02445 |
| H     | -0.95711 | 0.48697 | 0.63988 |
| H     | -5.49549 | 0.36737 | -0.93694 |
| H     | 5.64447 | 1.22220 | -0.43211 |
| H     | -4.21257 | 2.38648 | -0.73022 |
| C     | -4.59714 | -2.18360 | -0.24650 |
| H     | -4.03894 | -2.91022 | -0.84028 |
| H     | -4.72111 | -2.60768 | 0.75283 |
| H     | -5.58585 | -2.05938 | -0.69195 |
| H     | -0.97524 | -1.82654 | 0.62313 |

Sum of electronic and thermal free energy: –921.06397 a.u.

WC G*•C (C₁)

| X     | Y     | Z     |
|-------|-------|-------|
| N     | -0.74362 | 0.20970 | 0.67433 |
| C     | -1.34320 | -0.90525 | 0.06414 |
| C     | -2.72438 | -0.63350 | -0.32704 |
| C     | -3.33888 | 0.59216 | 0.13540 |
| N     | -2.70285 | 1.62463 | 0.56432 |
| C     | -1.32604 | 1.38960 | 0.16742 |
| N     | -0.53988 | 2.48920 | 0.05580 |
| O     | -0.76719 | -1.97282 | -0.05322 |
| N   | -3.68252 | -1.55340 | -0.64614 |
|-----|----------|----------|----------|
| C   | -4.81925 | -0.95355 | -0.41218 |
| N   | -4.69250 | 0.31087  | 0.10785  |
| H   | -1.00980 | 3.31393  | -0.23319 |
| H   | -5.43283 | 0.95650  | 0.32256  |
| H   | -5.79103 | -1.38791 | -0.60055 |
| C   | 2.85900  | 1.08064  | -0.23239 |
| N   | 2.16568  | -0.05662 | 0.02115  |
| C   | 2.79022  | -1.21847 | 0.16160  |
| C   | 4.22526  | -1.33130 | 0.10361  |
| C   | 4.91351  | -0.19456 | -0.12636 |
| N   | 4.25706  | 0.97955  | -0.29350 |
| O   | 2.34558  | 2.17666  | -0.40947 |
| N   | 2.03979  | -2.30390 | 0.36856  |
| H   | 4.75117  | 1.83862  | -0.47704 |
| H   | 5.99372  | -0.16055 | -0.19252 |
| H   | 4.72812  | -2.27856 | 0.23338  |
| H   | 2.46630  | -3.21096 | 0.42809  |
| H   | 1.02011  | -2.23338 | 0.28800  |
| H   | 0.43391  | 2.38076  | -0.23008 |
| H   | 0.27956  | 0.17310  | 0.63655  |

Sum of electronic and thermal free energy: –937.10615 a.u.

\( N_{\text{im}} = 0 \)

| WC \([G \rightarrow C]^* \) \((C_i)\) | X       | Y       | Z       |
|-------------------------------------|---------|---------|---------|
| O                                  | 2.29853 | 2.20235 | -0.07875|
| C                                  | 2.90130 | 1.12915 | -0.05198|
| N                                  | 2.25801 | -0.06870| 0.06160 |
| C                                  | 2.91565 | -1.30577| 0.11733 |
| C                                  | 4.28893 | -1.31682| 0.02643 |
| C                                  | 4.99098 | -0.13334| -0.10945|
| N                                  | 4.25644 | 1.06804 | -0.13500|
| N                                  | 2.10890 | -2.39531| 0.36608 |
| O                                  | -0.88195| -2.04765| -0.18148|
| C                                  | -1.41080| -0.94928| -0.09313|
| N                                  | -0.68534| 0.21251 | -0.01062|
| C                                  | -1.27001| 1.39324 | 0.07122 |
| N                                  | -2.63921| 1.66063 | 0.11308 |
| C                                  | -3.34088| 0.57237 | 0.04260 |
| C                                  | -2.86257| -0.76391| -0.06561|
| N                                  | -0.51428| 2.48870 | 0.12296 |
| N                                  | -3.88489| -1.63875| -0.11828|
| C                                  | -4.97142| -0.89126| -0.04542|
| N                                  | -4.71489| 0.44123 | 0.05303 |
| H                                  | 2.52792 | -3.28634| 0.16078 |
| H                                  | -0.97422| 3.38010 | 0.18120 |
| H                                  | 1.14207 | -2.33261| 0.07924 |
| H                                  | 0.50716 | 2.42181 | 0.06468 |
| H                                  | 4.82235 | -2.25593| 0.06937 |
| H                                  | 6.05833 | -0.05303| -0.21183|
| H                                  | 1.23955 | -0.02501| 0.07490 |
| H                                  | -5.97945| -1.27751| -0.06051|
| H                                  | -5.38262| 1.19131 | 0.11908 |
| H                                  | 4.71819 | 1.95687 | -0.21002|

Sum of electronic and thermal free energy: –937.15689 a.u.

\( N_{\text{im}} = 0 \)