Changes in the mesocarp of *Annona cherimola* Mill. ‘Madeira’ during postharvest ripening

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**Abstract**

Physicochemical changes during postharvest ripening of cherimoya (*Annona cherimola* Mill. ‘Madeira’), were investigated to follow the principal modifications occurring during this process and to determine nutritional value. Fruit harvested at the mature green stage were analyzed during ripening using standard methods. Significant (*P* < 0.05) changes in chlorophyll, starch, titratable acidity, total free sugars and uronic acids were obtained, but no significant changes were found in ash, protein, lignin and lipid contents during ripening. The most obvious changes were chlorophyll degradation, an accentuated decrease of starch and an increase in total free sugars, with glucose the predominant sugar in the mesocarp, as revealed by GC analyses. Firmness loss was mainly attributed to depolymerization of pectin and lipid deterioration rather than hemicellulose degradation. Results also showed that the cherimoya variety evaluated in this study is a good source of minerals (mainly potassium), palmitic acid, linoleic acid, α-linolenic acid and sitosterol.

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**1. Introduction**

The consumption of tropical and subtropical fruit is increasing worldwide (FAO, 2012), with consumer demand for new tastes. The cherimoya (*Annona cherimola* Mill.) is a soft subtropical fruit well adapted to the edaphoclimatic conditions of Madeira Island. The main cultivars of cherimoya produced in Madeira have been analyzed in a previous study (Caldeira et al., 1995), which referred to ‘Madeira’ as superior in commercial and organoleptic characteristics, giving it a high potential for commercialization in national and international markets. Fruit are harvested when they have turned from pale green to yellow-green, the areas between the few small conical protuberances have filled out or when skin gives a little to touch. Thus, the tender skin and the short shelf-life, 5–7 days, makes the fruit vulnerable to physical injuries after harvesting, during handling, transport and marketing, restricting its commercialization. These facts imply that producers have to harvest before fruit ripening. To extend the postharvest life of this fruit, they can be stored either at low temperature, which usually leads to chilling injury below 10 °C (Alique et al., 1994), depending on the cultivar, or under controlled modified atmosphere (Alique and Oliveira, 1994). Although this is a common practice nowadays, it results in severe loss of fruit quality in relation to texture, taste and flavor (Pareek et al., 2011).

In general, fruit ripening is complex and the mechanisms by which fruit soften during this process are unclear and subject to speculation. During ripening some modifications in the chemical composition, either by enzymatic or non-enzymatic processes (Brummell, 2006), lead to remarkable changes in fragrance, flavor and a decrease in pulp firmness. Decrease in fruit firmness is due, at least in part, to the disassembly of the cell walls which are a complex intertwining network containing cellulose/hemicelluloses embedded in an amorphous gelatinous matrix formed mainly by pectins and stabilized by (glyco) proteins and phenolics. For many fruit species, postharvest ripening is accompanied by an increase in pectin solubility and loss of neutral sugars mainly galactose, arabinose and mannose which is related to depolymerization of pectins (Manrique and Lajolo, 2004). Although the modifications of cell wall polysaccharides seem to be widespread among several fruit species, variations in cell wall composition could lead to differences in the softening process depending on the species.

In cherimoya, some investigations have been undertaken on fruit quality parameters to ascertain the changes taking place during postharvest ripening (Martínez et al., 1993; Gutiérrez et al., 2005; Goji et al., 2007). However, as the chemical composition depends on the cultivar, environmental conditions and also on the ripe stage of the fruit, the present study aimed to evaluate the main chemical changes that occurs on the mesocarp of cherimoya ‘Madeira’ which are essentially related to fruit quality. This
knowledge is important to understand the nutritional value and commercialization potential of this fruit and may help to interfere with the process of ripening in order to extend postharvest life.

2. Materials and methods

2.1. Sample preparation and physical parameters

Cherimoya (A. cherimola Mill. ‘Madeira’) fruit were harvested (early January) from trees in a commercial orchard in Faial (Madeira Island, Portugal). Fruit at mature green stage with no evidence of physical or pathological injuries were selected. Each fruit was carefully washed with sodium hypochlorite (2%) to remove potential contaminants. Fruit were weighed at the beginning of the experiment and at the end of each storage period using a digital balance. The difference between initial and final fruit weight was considered as weight loss during that storage period and was expressed in percentage. The density of each cherimoya was determined by the ratio of weight to volume displaced after fruit immersion into a measuring cylinder. Afterwards, lots of five fruit were randomly selected and stored at room temperature (20–22 °C) in a dimly-lit place. Every day after harvest, fruit were randomly chosen for physicochemical analyses. Fruit firmness was determined after removing the skin on two opposite sides at the middle of each fruit using a pressure-testing instrument (Model FT 327) fitted with an 11.3 mm cylindrical plunger. The force required to penetrate into the flesh was expressed in N. Penetration was carefully performed to avoid nearby seeds. Immediately, fruit were peeled (green peel was fully discarded), sliced, quick-frozen in liquid nitrogen and stored at −80 °C. From each sample a fresh slice was used to measure fruit water content using a humidity balance (Giberti-Eurotherm) at 105 °C. Frozen mesocarp samples were lyophilized, milled and stored in the dark under vacuum for further analyses. In storage the average humidity was approximately 5%.

2.2. Chemical analyses

To determine chlorophyll content, 1 g of sample and 5 mL of acetone at 80% (v/v) were homogenized with a vortex and submitted to ultrasound extraction for 5 min at 25 °C. Samples were then centrifuged (Biofuge Stratos, Heraeus) at 1600 rpm for 5 min. The supernatant was placed in a quartz cell and its absorbance was measured by an ultraviolet-visible spectrometer (UV-2401, Shimadzu), at 663 nm (chlorophyll a) and at 646 nm (chlorophyll b) using 80% acetone as blank solution. Chlorophyll analysis was performed in duplicate in a dimly light room and its content was determined according to the equation of Lichtenhaler (1987). Unless otherwise stated, chemical analyses were performed according to AOAC (2000). Ash content was determined by complete incineration of 1 g of mesocarp sample in a Nabert therm furnace at 600 °C for 6 h. Crude protein content was calculated by converting the nitrogen content (N × 6.4), determined by the Kjeldahl method in a Kjeldahl Selecta Alcodest still. The lignin content was determined using the Klasson method (T 204 om-88). Titratable acidity was determined twice and the results expressed as citric acid equivalents. The content of uronic acids was obtained based on the method of m-phenylenediamine with the galacturonic acid as standard. The iodine colorimetric method was used to measure starch content. Total free sugars (sucrose and reducing sugars) were determined according to Dubois et al. (1956), using glucose as the standard. The sugars monomers were determined by acid hydrolysis (Blakeney et al., 1983) followed by gas chromatography (GC), using a HP 5890 chromatograph equipped with a fused silica capillary column J&W DB-225 (30 m × 0.25 mm i.d.; 0.15 μm film thickness). Before sample injection, calibration curves for rhamnose, fucose, arabinose, xylose, mannose, galactose and glucose were obtained using high purity commercial standards.

2.3. Lipophilic extracts

Milled mesocarp samples were extracted by Soxhlet with dichloromethane during 6 h. Dichloromethane was selected as a specific solvent for lipophilic extractives isolation for analytical purposes. The solvent was evaporated to dryness and the amount of extracts determined by gravimetry. The lipophilic extractives were identified and quantified by gas chromatography–mass spectrometry (GC–MS) as described by Oliveira et al. (2006). Briefly, 20 mg of each dried extract with a measured amount of internal standard was dissolved in 250 μL of pyridine. After the addition of 250 μL of bis(trimethylsilyl)trifluoroacetamide and 50 μL of trimethylchlorosilane, the mixture stayed at 70 °C for 30 min. GC–MS analyses were performed using the Agilent 6890N gas chromatography coupled to a 5975 Agilent mass selective detector, equipped with a DB-1 column (J&W: 30 m × 0.25 mm i.d.; thickness, 0.25 μm), using the conditions described previously (Oliveira et al., 2006). Components were identified based on the comparison of their spectra with two spectral libraries (NIST/EPA/NIH Mass Spectral Database, US), the retention times and, in some cases by comparing their fragmentation profiles with published data. For quantitative analysis, GC–MS was calibrated with pure reference compounds, representative of the major lipophilic extractives components (namely hexadecanoic acid, 1-eicosanol, 16-hydroxyhexadecanoic acid, ferulic acid and stigmasterol), relative to tetracosane used as internal standards. For each sample two injections were performed and results represent the average of six injections. Between injections the variation was less than 6%.

2.4. Statistical analysis

All chemical analysis and fractionation experiments were carried out at least twice and the results presented are the average of the values obtained for each fruit with a standard deviation lower than 6%. The experimental data were statistically analyzed by one factor analysis of variance (ANOVA) to determine its significance at P < 0.05, using the SPSS (Statistical Package for Social Science) version 15.0 for Windows.

3. Results and discussion

3.1. Physical parameters

The density of cherimoya fruit ranged from 0.88 g cm⁻³ to 1.11 g cm⁻³, without any important variation during the ripening period. Weight loss increased steadily during storage (Fig. 1) and at the end of the experiment fruit had lost around 9% in weight. No statistical relationship was found between the density and the fruit weight. Fruit weight loss during ripening was also reported by others in cherimoya (Alique et al., 1994), custard apple (Prasanna et al., 2000) and sour sop (Lima et al., 2006), being mainly attributed to water loss through respiration, transpiration and ripening. Water content in ‘Madeira’ cherimoyas ranged between 73 and 83% similarly to what has been reported for custard apple, cherimoya, sour sop and sugar apple (Pinto et al., 2005).

Penetration force used to evaluate cherimoya softening was higher than 63.6 N on the first two days after fruit harvest and decreased sharply at day 4, reaching an average of 2.7 N at the end of storage (Fig. 1). At day 5 the fruit were over-ripe showing a softer texture that affects the quality for marketing. This enhanced rate of softening is in agreement with previous studies (Pareek et al., 2000).
and can be attributed to several physiological processes that proceed in the course of ripening.

### 3.2. Chemical composition

After fruit harvest, the content of chlorophyll slightly increased on the second day and decreased significantly \((P<0.05)\) on the third day of storage (Fig. 2). In fleshy fruit, chlorophyll breakdown is induced at the onset of ripening and may serve to indicate the degree of ripening (Barry, 2009). Chlorophyll degradation during fruit ripening might be due in part to chlorophyllase activity (Almela et al., 2000) which result in chlorophyll catabolites that contribute to alter aroma profiles (Barry, 2009) and might play a role as antioxidants (Müller et al., 2007).

The ash content was similar among fruit and during ripening \((P>0.05)\) (Table 1). Mineral composition showed that ‘Madeira’ cherimoyas are a good source of potassium but have moderate amounts of phosphorus, magnesium and calcium (Table 2). These results are consistent with those found by Leterme et al. (2006) for *A. cherimola*.

Total protein content decreased during the postharvest ripening (Table 1), although without significant differences \((P>0.05)\). Proteins provide amino acids, which are used as precursors of volatile compounds that are formed during the ripening process, giving typical aroma to fruit. Additionally, it can also provide nitrogen that might be reassimilated to form nitrogenous compounds such as polyamines that increase during the ripening of cherimoya (Escribano and Merodio, 1994). However, storage of *A. cherimola*

‘Fino de Jete’ at 20 °C showed an increase in the levels of proteins (Maldonado et al., 2002). This discrepancy might result from the postharvest physiology of both cultivars, the climatic conditions during the development and growth of cultivars, or the method used to measure the protein content.

Lignin content decreased slightly from the first to the second day in cherimoya fruit stored at room temperature (Table 1). From the third day onwards the increase in lignin was slow but not significant \((P>0.05)\). Lignin accumulation during cherimoya maturation might represent the presence of sclerenchyma as mesocarp encloses scleresids that contain substantial quantities of lignin. The accumulation of lignin during the ripening process of cherimoya ‘Madeira’ fruit was contrary to what was observed in cherimoya ‘Fino de Jete’ (Assis et al., 2001) and *Psidium cattleyanum* (Galho et al., 2007).

In unripe cherimoya, starch accounts for about 4% of the fresh pulp weight (Fig. 3). The starch content declined sharply during storage at room temperature and four days after harvest was negligible. A similar trend in starch breakdown during cherimoya ripening has previously been reported (Martínez et al., 1993; Gutiérrez et al., 1994). Statistical differences were found only on the second and third days of ripening \((P<0.05)\). The fast hydrolysis of starch was not paralleled by the accumulation rate of total free sugars. Therefore, besides starch being partially converted to free sugars, it is also transformed into organic acids, which is consistent with both increases in total free sugars (Fig. 3) and titratable acidity (Table 1).

Total sugars began to accumulate in unripe fruit and increased by almost 90% at the end of storage (day 5) when fruit were ripe (Fig. 3). However, the rise in free sugars was only significant between the first and second days \((P<0.05)\). The GC analyses of neutral sugars revealed that the predominant sugar in the mesocarp of cherimoya ‘Madeira’ was glucose \((ca. 90\%)\). Altogether, the levels of free sugars present indicate that ‘Madeira’ is sweeter than other previously studied cultivars (Caldeira et al., 1995). The absence of monomers such as xylose, arabinose, rhamnose, amongst others, allow us to conclude that there was not a considerable degradation of hemicellulosics during the ripening process, therefore the loss of firmness of the fruit perhaps cannot be connected to the alterations of these polymeric sugars.

The titratable acidity showed a considerable increase between days 2 and 3 after postharvest (Table 1). This increase in acidity was significant \((P<0.05)\) and can be ascribed to the production of organic acids, mainly malic acid (Alique et al., 1994), which contributes to the fruit aroma and influence perception of sweetness. At the end of the experiment the titratable acidity was two-fold higher than that measured on the first day. An increase in acidity was also reported in *A. squamosa* L. (Prasanna et al., 2000).
Table 1
Chemical composition of 'Madeira' cherimoyas during postharvest ripening. Fruit were stored at 20–22 °C after harvest.

| Chemical composition (g 100 g⁻¹ fresh pulp) | Days after harvest |
|---------------------------------------------|-------------------|
|                                             | 1                 | 2                 | 3                 | 4                 | 5                 |
| Ash                                         | 0.884 ± 0.174     | 0.880 ± 0.220     | 0.885 ± 0.229     | 0.892 ± 0.179     | 0.836 ± 0.186     |
| Protein                                     | 1.523 ± 0.191     | 1.469 ± 0.264     | 1.511 ± 0.159     | 1.250 ± 0.308     | 1.196 ± 0.130     |
| Lignin                                      | 2.150 ± 0.526     | 1.969 ± 0.222     | 2.695 ± 0.180     | 2.759 ± 0.272     | 3.105 ± 0.254     |
| Titratable acidity                          | 0.113 ± 0.024     | 0.101 ± 0.009     | 0.207 ± 0.025     | 0.227 ± 0.033     | 0.256 ± 0.019     |
| Uronic acids                                | 0.065 ± 0.010     | 0.141 ± 0.022     | 0.186 ± 0.035     | 0.307 ± 0.004     | 0.324 ± 0.032     |
| Lipophilic extractives                      | 0.324 ± 0.032     | 0.312 ± 0.072     | 0.347 ± 0.072     | 0.225 ± 0.034     | 0.221 ± 0.017     |

Table 2
Mineral composition of ‘Madeira’ cherimoyas during postharvest ripening.

| Days after harvest | Mineral composition (mg 100 g⁻¹) |
|--------------------|---------------------------------|
|                    | K | P  | Ca     | Na  | Mg | Fe | Cu  | Zn | Mn |
| 1                  | 406.8 | 17.9 | 11.3 | 15.9 | 177.0 | 0.12 | 0.06 | 0.13 | 0.31 |
| 3                  | 379.5 | 18.8 | 9.0  | 6.4  | 16.2 | 0.12 | 0.10 | 0.23 | 0.12 |
| 4                  | 411.0 | 35.6 | 31.6 | 23.4 | 29.8 | 0.26 | 0.14 | 0.23 | 0.40 |
| 5                  | 379.1 | 57.8 | 49.2 | 32.5 | 45.5 | 0.79 | 0.22 | 0.44 | 0.40 |

Uronic acids increased during postharvest ripening with a significant rise (P<0.05) on the second and fourth days (Table 1). Uronic acid solubilization is coupled with the activity of pectolytic enzymes such as polygalacturonase (Sánchez et al., 1998) and β-galactosidase that increase continuously in A. muricata L. (Lima et al., 2006) during postharvest storage or pectin methylesterase believed to be involved in degradation of pectin cell wall components at the initial stages of ripening (Guadarrama and Andrade, 2012). Pectin solubilization has been observed in other fleshy fruit species as Carica papaya L. (Manrique and Lajolo, 2004) and Prunus persica L. Batsch (Ghiani et al., 2011) contributing to the softening mechanism.

3.3. Characterization of the lipophilic extractives

The lipid metabolism changed during fruit ripening as the lipophilic extractives decreased from 0.324 g per 100 g of fresh pulp in the first day to 0.221 g per 100 g of fresh pulp in the fourth day after harvest (Table 1), yet this reduction was not statistically different (P>0.05). Lipid composition of fruit, especially fatty acids (α-linolenic and linoleic acids), is important due to its health potential as it prevents, delay, or treat chronic and acute diseases, such as cancer, cardiovascular diseases, osteoporosis, cholesterol and immune disorders (Chen et al., 2013).

Table 3
Compounds (mg per kg of dry material) identified in the dichloromethane extracts of ‘Madeira’ cherimoyas during postharvest ripening.

| Compound (mg kg⁻¹, dry material) | Days after harvest |
|----------------------------------|-------------------|
|                                  | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 |
| Fatty acids                      | 2770 | 2130 | 1771 | 1116 | 868 |
| Saturated                        | 1467 | 1180 | 1144 | 626 | 573 |
| Pentadecanoic acid               | 7 | 1 | 2 | 2 | n.d. |
| Palmitic acid                    | 866 | 699 | 567 | 322 | 269 |
| Heptadecanoic acid               | 29 | 21 | 22 | 11 | 9 |
| Octadecanoic acid                | 142 | 94 | 98 | 48 | 52 |
| Nonadecanoic acid                | n.d. | 5 | n.d. | n.d. | 2 |
| Hexadecanoic acid                | 45 | 37 | 45 | 29 | 24 |
| Eicosanoic acid                  | 5 | n.d. | n.d. | n.d. | n.d. |
| Docosanoic acid                  | 34 | 26 | 15 | 8 | 3 |
| Tricosanoic acid                 | 115 | 119 | 129 | 54 | 77 |
| Tetracosanoic acid               | 254 | 23 | 21 | n.d. | n.d. |
| Pentacosanoic acid               | n.d. | 155 | 245 | 152 | 137 |
| Unsaturated                      | 1273 | 950 | 627 | 490 | 295 |
| Palmitelaidic acid               | 8 | 6 | 5 | 2 | n.d. |
| Oleic acid (cis or trans)        | 180 | 238 | 146 | 114 | 135 |
| Linoleic acid                    | 683 | 423 | 328 | 183 | 95 |
| α-linolenic acid                 | 502 | 283 | 148 | 191 | 65 |
| Sterols                          | 1431 | 1457 | 1292 | 876 | 1216 |
| Campesterol                      | 154 | 156 | 142 | 96 | 134 |
| Stigmasterol                     | 238 | 240 | 248 | 162 | 224 |
| Sotosterol                       | 934 | 997 | 902 | 564 | 788 |
| Isoucosenol or ovaenesterol      | 105 | 64 | n.d. | 54 | 70 |
| Long chain aliphatic alcohols    | 49 | 65 | 62 | 120 | 44 |
| 1-Octadecanol                    | n.d. | 2 | 5 | 83 | n.d. |
| 1-Docosanol                      | 15 | 8 | 13 | 3 | 5 |
| 1-Tetracosanol                   | 25 | 55 | 40 | 34 | 34 |
| 1-Octacosanol                    | 9 | n.d. | 4 | n.d. | 5 |
| δ-Tocopherol                     | 34 | 12 | 32 | 5 | n.d. |

n.d.: not detected.
The analyses by GC–MS revealed that the quantities of lipophilic extracts varied as fruit ripen and were mainly formed by fatty acids and sterols (Table 3). Fatty acids accounted for 41–65% of the total extractives. Among the saturated fatty acids palmitic, pentacosanoic, tricosanoic and octadecanoic acids were predominant (Table 3). α-Linolenic acid (omega-3) and linoleic acid (omega-6) represented 8–18% and 11–25%, respectively, of the total fatty acids content. During cherimoya ripening the saturated and unsaturated fatty acids decreased from 1497 to 573 mg kg⁻¹ and 1273 to 295 mg kg⁻¹ of pulp, respectively. This reduction could be due to the degradative lipolytic enzymes (Palijathy and Thompson, 1987) that are able to degrade endogenous lipids, thus changing the lipid bilayer and accelerating permeability in senescing membranes. The chemical changes in the lipid bilayer might include the loss of phospholipids and fatty acids, and an increase in the sterol: fatty acid ratio. In the present study the sterol: fatty acid ratio increased from 0.52 to 1.40, values similar to those obtained for 'Fino de Jete' cherimoyas (Gutiérrez et al., 2005).

The sterols were the second largest family in the lipophilic extractives comprising 33–57% of the total extracts. Sitosterol (65–70% of total sterols), stigmasterol (17–19%) and campesterol (about 11%) were the prevalent sterols present in cherimoya ‘Madeira’. The sterol composition of this cultivar can contribute to the intake of natural phytosterols in the human diets, which appear to be a practical and safe option for reducing cholesterol levels in the population (Quilèz et al., 2003).

The percentage of long chain aliphatic alcohols in the lipophilic extract ranged from 1.1 to 5.7% 1-tetracosanol being the most predominant (28–85% of the total of this family) followed by 1-docosanol, with values ranging between 3 and 31% (Table 3). 1-Octacosan-1-ol increased significantly at day 4, accounting for 69% of the total long chain aliphatic alcohols. It may be assumed that this increase at day 4 might coincide with the production of ethylene, as this compound is necessary for the coordination and completion of ripening in climacteric fruit (Giovannoni, 2001). Reports on the role of long chain aliphatic alcohols in human health suggest a decrease in the low-density lipoprotein cholesterol and an increase in the high-density lipoprotein cholesterol (Hargrove et al., 2004).

In this study, only δ-tocopherol was detected and decreased with fruit maturation, being almost absent around day 4. Like cherimoya, this phenomenon has been described for other climacteric fruit such as tomatoes (Abushita et al., 1997) and grapes (El-Shami et al., 2001). The reduction of δ-tocopherol might be related to the transformation into α-tocopherol (Schultz, 1990), although the α-tocopherol was not detected in any of the samples analyzed. Despite the low content of tocopherol in cherimoya, their consumption together with other plant-derived foods provide a significant source of vitamin E, which has been associated with the prevention of cardiovascular diseases, cancer, inflammatory diseases, neurological disorders, cataract and age-related macular degeneration, as well as to the maintenance of the immune system (Bramley et al., 2000).

### 4. Conclusion

The results showed that the quality attributes and nutritional contents of ‘Madeira’ cultivar changed significantly during postharvest ripening, with substantial changes occurring on the third day. The outcomes of this study emphasize ‘Madeira’ as an excellent nutrient resource. The lipophilic extractives data are a relevant contribution for the value of this cherimoya cultivar as source of highly valuable phytochemicals such as omega-3 and omega-6 fatty acids, and sterols, which are well known for their beneficial impacts on health.

Considering the importance of texture in consumer acceptability of fruit quality further studies in fruit physiology should be explored for desired postharvest fruit handling or suppressing expression of specific enzymes that contribute to softening.

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