Carbon related defects in irradiated silicon revisited

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Electronic structure calculations employing hybrid functionals are used to gain insight into the interaction of carbon (C) atoms, oxygen (O) interstitials, and self-interstitials in silicon (Si). We calculate the formation energies of the C related defects C(SiI), C(O), C(C), and C(OI)Si with respect to the Fermi energy for all possible charge states. The C(SiI)1+ state dominates in almost the whole Fermi energy range. The unpaired electron in the C(O)1+ state is mainly localized on the C interstitial so that spin polarization is able to lower the total energy. The three known atomic configurations of the C(C) pair are reproduced and it is demonstrated that hybrid functionals yield an improved energetic order for both the A and B-types as compared to previous theoretical studies. Different structures of the C(OI)Si cluster result for positive charge states in dramatically distinct electronic states around the Fermi energy and formation energies.

Silicon (Si) is the basic material for numerous microelectronic, photovoltaic and sensor devices. Its electronic properties are known to be significantly affected by the presence of impurities and defects, which play an increasingly important role with the miniaturization of devices. Carbon (C), along with oxygen (O), is the most common and important impurity in Si. C is being incorporated in the Si lattice inadvertently during the growth. It mainly originates from the poly-crystalline starting material, from graphitic components in the equipment or/and from gaseous contaminants developed during the growth process. C is isovalent with Si and occupies substitutional sites (C_s), which are electrically neutral. Nevertheless, the introduction of C in the lattice leads to local strain, due to its smaller size as compared to Si. Its presence is evidenced in infrared spectra by a localized vibrational mode at 607 cm^-1. Conversely, O impurities (which are likewise electrically neutral) occupy interstitial sites (O_i) within the Si lattice. Their presence is evidenced in infrared spectra by a number of localized vibrational modes, the most important among them at 1106 cm^-1. Both impurities are highly electronegative and chemically very reactive. They readily form bonds with Si atoms, with other defects and impurities, and between them. For instance in as-grown Cz-Si, modes at 589 cm^-1, 640 cm^-1, and 690 cm^-1 have been correlated with C-O complexes, in particular C_s-O_i pairs.

Upon irradiation, for instance with electrons of 1–2 MeV energy, vacancies and Si interstitial (Si_I) atoms are initially formed. Importantly, in spite of various suggestions in the past, there is no adequate experimental evidence for any interaction between vacancies and C atoms. However, it is established that most of the Si_I are readily trapped by C_s, which are pushed to interstitial sites according to the Watkins displacement reaction C_s + Si_I → C. Two localized vibrational modes at 922 and 932 cm^-1 have been correlated with this defect in low temperature irradiated Si. C also introduces in-gap states. The defect is unstable at room temperature and upon migration interacts promptly with O_i and C_s to form C(O)_s and C(C)_s pairs. At least six localized vibrational modes have been correlated with the C(O)_s defect. The best known appears at 865 cm^-1 and, as the strongest, is used for calculating the defect concentration by a calibration coefficient of 1.1 \times 10^{17} cm^-2. The C(C)_s defect is bistable and at least eleven localized vibrational modes are correlated with it in both configurations. All of them can be detected by low temperature measurements and the one at 544 cm^-1 can be seen at room temperatures. For the calculation of the defect concentration a calibration coefficient of 1.5 \times 10^{17} cm^-2 is used. Notably, a third configuration, the <100> C-C dumbbell, was proposed by Liu et al. combining theoretical and experimental work. It was calculated that this configuration energetically is lower than the other bistable structures. Importantly, both the C(O)_s and C(C)_s defects introduce states within the Si band gap, affecting the efficiency of the corresponding devices. Thus, numerous experimental and theoretical studies have been performed to investigate the structure, properties, and behavior.

C-related defects are technologically particularly important. For example the C(C)_s defect is utilized to improve the performance of Si optical emitters. Additionally, the C_s, C(O)_s, and C(C)_s defects usually act in the course of...
Figure 1 | Structures of (a) $C_i$ (Si), (b) $C_iO$, (c) $C_iO_2(Si)_i$, (d) $C_iO_2(Si)_b$, (e) A-type $C_iC_o$, (f) B-type $C_iC_o$, and (g) C-type $C_iC_o$. Big blue spheres are Si atoms, medium yellow spheres C atoms, and small red spheres O atoms. The SiI atom in Ci (Si), the Si atom connecting two C atoms in A-CiCs and B-CiCs, and the two Si atoms that show a significant position change between the two CiOi(SiI) structures are marked in gray. The breaking and forming bonds during the transition between A-CiCs and B-CiCs are highlighted by arrows.
Figure 2 | Total DOSs of pristine Si (gray) and the $C_1(Si_I)$ defect (red). The states left of the dotted line are occupied.

Figure 3 | Formation energies of the (a) $C_1(Si_I)$, (b) $C_2O_I$, (c) $C_2O_I(Si_I)_a$, (d) $C_2O_I(Si_I)_b$, (e) A-type $C_2C_s$, (f) B-type $C_2C_s$, and (g) C-type $C_3C_s$ defects with respect to the Fermi energy.
irradiation as nucleation centers for self-interstitials and complexes such as C\textsubscript{i}(Si\textsubscript{I}), C\textsubscript{i}O\textsubscript{i}(Si\textsubscript{I}) and C\textsubscript{i}Cs(Si\textsubscript{I})\textsuperscript{25–28}. Notably, it was previously concluded, based on numerical simulations, that the capture radius of self-interstitials at C\textsubscript{i}O\textsubscript{i} is more than 3 times larger than at C\textsubscript{s}\textsuperscript{4}.

Given the significance of self-interstitials for the Si properties, any contribution towards the understanding of their defect processes is crucial. For example, C\textsubscript{i}(Si\textsubscript{I}) that has a significant role in the C\textsubscript{ag} aggregation processes in Si can impact the transient enhanced diffusion of boron\textsuperscript{27}. Two infrared bands at 953 cm\textsuperscript{-1} and 966 cm\textsuperscript{-1} have been attributed to the C\textsubscript{i}(Si\textsubscript{I}) complex, while those at 940 cm\textsuperscript{-1} and 1024 cm\textsuperscript{-1} belong to the C\textsubscript{i}O\textsubscript{i}(Si\textsubscript{I}) complex.

In the present study, we use hybrid density functional theory to characterize the critical C-related defects. In particular, we investigate the binding energies of the basic C\textsubscript{i}(Si\textsubscript{I}), C\textsubscript{i}O\textsubscript{i}, C\textsubscript{i}Cs, and C\textsubscript{i}O\textsubscript{i}(Si\textsubscript{I})\textsubscript{a} defects as well as the formation energies with respect to the Fermi energy for all possible charge states.

### Results and discussion

#### Isolated defects

Before considering both defect pairs and clusters we investigated their constituent components C\textsubscript{o}, C\textsubscript{i}, O\textsubscript{i}, and Si\textsubscript{I}. We reproduce the structure of O\textsubscript{i} according to Ref. 29 and the references therein. The O interstitial is bonded to two Si atoms forming a non-linear Si-O-Si pseudo-molecule. The two Si-O bond lengths are 1.63 Å and 1.64 Å, and the Si-O-Si bond angle is 151°. The charge neutral states of C\textsubscript{o} and O\textsubscript{i} dominate in the whole Fermi energy range, where the formation energies are 2.22 eV and 1.95 eV, respectively. These values are in good agreement with the heat of solution of substitutional C (2.30 eV) and interstitial O (1.65 ± 0.15 eV) in C\textsubscript{z}-Si, as determined by Bean and Newman\textsuperscript{30}. The <1 1 0> dumbbell C\textsubscript{i} structure is also implemented in present study. It is the most favorable configuration according to Zirkelbach et al\textsuperscript{31}. The dumbbell of C and Si atoms occupies a Si site with a bond distance of 1.75 Å. In the +1 charge state the unpaired electron is mainly located on the C interstitial, whereas in the −1 charge state the Si atom in the dumbbell traps the wave function of the electron more than other Si atoms because it loses valence charge to the C atom due to the higher electronegativity. Spin polarized HSE calculations show a lowering of the total energy for the charge +1 state by 0.27 eV as compared to spin degenerated calculations. C\textsubscript{i} favors the +2 charge state at low Fermi energy (the formation energy is 3.42 eV for a Fermi energy of 0 eV), the +1 charge state in the middle of the band gap, and the 0 charge state with a formation energy of 4.50 eV in the high Fermi energy range. This value is larger than the 3.72 eV calculated within the general gradient approximation\textsuperscript{31}. The charge transition levels amount to 0.23 eV (\textsuperscript{1}1/\textsuperscript{1}1) and 0.85 eV (\textsuperscript{0}/\textsuperscript{2}2).

Table I | Transition levels (eV) between charge states for the C\textsubscript{i}(Si\textsubscript{I}), C\textsubscript{i}O\textsubscript{i}, C\textsubscript{i}Cs, and C\textsubscript{i}O\textsubscript{i}(Si\textsubscript{I})\textsubscript{a} defects

| Transition level | C\textsubscript{i}(Si\textsubscript{I}) | C\textsubscript{i}O\textsubscript{i} | A-C\textsubscript{i}C\textsubscript{s} | B-C\textsubscript{i}C\textsubscript{s} | C-C\textsubscript{i}C\textsubscript{s} | C\textsubscript{i}O\textsubscript{i}(Si\textsubscript{I})\textsubscript{a} |
|-----------------|------------------|------------------|------------------|------------------|------------------|------------------|
| (+ +/)          | 0.99             | 0.23             | ...              | ...              | 0.16             | 0.49             |
| (+ +/0)         | 1.01             | 0.46             | 0.16             | 0.12             | 0.06             | 0.60             |
| (+/0)           | 1.04             | 0.70             | 0.39             | 0.25             | ...              | 0.72             |
| (0/−)           | ...              | ...              | ...              | ...              | 1.05             | ...              |
| (0/−)           | ...              | ...              | ...              | ...              | ...              | 0.91             |
| (+/−)           | ...              | 0.73             | 0.74             | 0.50             | ...              | ...              |
| (+/−)           | ...              | 0.93             | 1.03             | 0.59             | ...              | ...              |
| (+−/−)          | 0.16             | 0.99             | 0.46             | 0.49             | 0.39             | 1.02             |
| (+−/−)          | 0.68             | 0.77             | 0.48             | ...              | ...              | ...              |

#### Defect pairs

The most stable configuration of the C\textsubscript{i}(Si\textsubscript{I}) defect is presented in Fig. 1(a). The Si-C dumbbell partially shares the...
interstitial site and the Si atoms surrounding the defect pair are slightly shifted off their original positions. The C atom is threefold coordinated. Total densities of states (DOSs) in Fig. 2 show that the defect states gradually become separated from the valence band as the system receives more and more positive charge. We note that the band gap of pristine Si is calculated to be 1.05 eV using HSE, which resembles the experimental value of 1.17 eV. The differences in the electronic structures arise from the distortion of the C-Si dumbbell and its neighboring atoms. The C-Si bond distances are reduced from 1.80 Å, 1.82 Å, and 1.82 Å in the charge neutral state to 1.77 Å, 1.79 Å, and 1.80 Å, respectively, in both positively charged states (which have the same C-Si bond distances). In addition, the orientation of the C-Si dumbbell is changed. The angle between the dumbbell and the [101] direction is 16°, 24°, and 29° for the 0, +1, and +2 charge state, respectively. The formation energies for the C\textsuperscript{i}(Si\textsuperscript{i}) defect, see Fig. 3(a), indicate that the +2 state is more favorable than other charge states up to high Fermi energy. The transition levels between charge states are given in Table I.

The structure of the C\textsubscript{O}\textsubscript{i} defect obtained in this work is consistent with the results of previous studies\textsuperscript{17,18}. The C and O interstitials form with Si a ring, see Fig. 1(b), where the three O-Si bond lengths amount to 1.75 Å, 1.78 Å, and 1.86 Å, and the three C-Si bond lengths to 1.77 Å, 1.78 Å, and 1.79 Å. The Si-O-Si and Si-C-Si angles are 89° and 92°, respectively. Total DOSs for this system are shown in Fig. 4. The electronic structure of the C\textsubscript{O}\textsubscript{i}\textsubscript{0} defect indicates that the valence bands and conduction bands occupy almost the same energy range as for pristine Si except for the fact that the defect states, which contain two electrons, shift up into the band gap. The center of the defect states is found about 0.35 eV above the valence band, which agrees well with the value of 0.38 eV as reported experimentally in Ref. 36.

Due to a pronounced DOS peak at the Fermi energy in the case of the C\textsubscript{O}\textsubscript{i}\textsuperscript{1} defect, HSE calculations with spin polarization are employed. We find a lowering of the total energy by 0.25 eV and a magnetic moment of 0.29 μ\textsubscript{B} on C. Magnetic moments on other atoms are negligible. This result agrees with the electron paramagnetic resonance study of the C\textsubscript{O}\textsubscript{i}\textsuperscript{1} defect in Ref. 16, which has found that the unpaired electron resides mainly on the C atom. Spin polarized total and partial DOSs for the C\textsubscript{O}\textsubscript{i}\textsuperscript{1} defect are plotted in Fig. 5. The Si atoms nearest to and farthest away from the defects are addressed as examples. For C and Si we show the sum of the s and p states because of the s-p hybridization. We find that the defect states are mainly due to the C interstitial, whereas the Si atoms around the defect give minor contributions. Hence, the C interstitial has a much larger magnetic moment than any other atom. As to be expected, Si atoms close to the C\textsubscript{O}\textsubscript{i} pair contribute more to the defect states than Si atoms with larger distance. The formation energy of the C\textsubscript{O}\textsubscript{i}\textsuperscript{1} as a function of the Fermi energy for different charge states is shown in Fig. 3(b), where the energy of the spin polarized calculation is used for C\textsubscript{O}\textsubscript{i}\textsuperscript{1}. Due to the energy gain by spin polarization, we find that the stable state transits from +2 to +1 and then to 0 charge as the Fermi energy increases.

Figures 1(e), (f), and (g) present the three stable configurations of the C\textsubscript{Cs} defect. The A- and B-types\textsuperscript{37} are established since long, whereas the C-type more recently has been demonstrated by first principles calculations to be the most stable configuration\textsuperscript{31}. The bond lengths around the defects obtained in the present study, compare Table II, are consistent with the results in the literature\textsuperscript{19,20,31}.

| A-C\textsubscript{Cs} | C-Si | C\textsubscript{Cs} | Si\textsubscript{2-C} | Si\textsubscript{2-C} |
|---|---|---|---|---|
| A-C\textsubscript{Cs} | 1.88, 1.99, 1.99, 2.03 | 1.75, 1.83, 1.83 | 1.75, 1.88 | 1.75, 1.88 |
| B-C\textsubscript{Cs} | C-Si | 1.85, 1.94, 2.01, 2.01 | 1.88, 1.96, 1.96, 2.04 | 1.85, 1.88 |
| C-C\textsubscript{Cs} | C-Si | 1.89 | 1.42 | 1.42 |

Figure 5 | Spin polarized (a) total and (b) partial DOSs of the C\textsubscript{O}\textsubscript{i} defect. Si(n) and Si(f) denote the Si atoms nearest to and farthest away from the defect pair, respectively. The states left of the dotted line are occupied.
C atoms as Si_{2C}. Compared with A-C_{iCs}, the Si_{2C}-Si bond breaks and a C_{i}-Si bond forms in B-C_{iCs}. Although the C interstitial is now fourfold coordinated and substitutes the position of Si, we keep the notation as C_{i} for systematic reasons. In C-C_{iCs}, the C-C pair along the <100> direction occupies a regular Si lattice site. Its bond length (1.42 Å) is shorter than in diamond or graphite. The fact that the four C-Si bonds (1.89 Å) are longer than the C-C bond implies that the C-Si interaction is weaker than the C-C dumbbell interaction.

The total DOSs for the 0 charge A-, B- and C-C_{iCs} defects are depicted in Fig. 6. In A-C_{iCs} the C_{i} has one dangling electron, leading to a peak below the Fermi energy. The 6.5 eV peak is mainly due to the localized unoccupied Si_{2C} states, which are caused by the two C nearest neighbors with higher electronegativity. Exactly the same situation is encountered in B-C_{iCs}. A large portion of the electron trapped by the C_{i} defect thus resides on Si_{2C}. Since C_{i} is fourfold coordinated in B-C_{iCs}, the Si_{2C} atom keeps more valence charge, see the pronounced DOS at 5.2 eV. Also, both C atoms are in fourfold coordination and therefore give no significant contribution to this peak. The fact that each C atom in C-C_{iCs} has a dangling electron is demonstrated by localized states at the Fermi energy in the spin degenerate DOS (not shown). The spin polarized DOS in Fig. 6 shows a clear spin splitting that results in magnetic moments on the C atoms. In the +1 and −1 states the wave function of the removed/trapped electron is shared by both C atoms.

The experimental energy differences between A- and B-C_{iCs} indicate that the A-type is more stable in the +1 and −1 charge states, whereas the B-type is more stable in the 0 charge state. For comparison, the results obtained by this and by previous studies are summarized in Table III. Those calculated by the PBEsol functional for the 0 charge state agree with the experiment only in the energetic order while the value is much larger. The HSE functional, on the other hand, yields results that agree with the experiment much better than previous theoretical findings, except for the −1 charge state where the deviation from the experimental value is 0.11 eV. This discrepancy between the theoretical and experimental energy differences may be due to the fact that for A- and B-C_{iCs}−1 the wave function of the unpaired electron is delocalized, as demonstrated by experiments, while such a scenario is not well described by the HSE functional. Spin polarized calculations are performed with both the PBEsol and HSE functionals. PBEsol gives almost the same total energies as the spin degenerate calculations, while HSE lowers the total energies of A-type C_{i}−1 and B-type C_{i}+1 and C_{i}C_{i}−1 by significant amounts of 0.22 eV, 0.02 eV, and 0.07 eV, respectively. Therefore, the energy difference between the two types is now −0.29 eV for the +1 charge state and 0.14 eV for the −1 charge state. In the case of A-C_{i}−1 the spin polarization delocalizes the occupied C_{i} DOS and shifts the unoccupied C_{i} DOS to higher energy. For B-type C_{i}C_{i}+1 and C_{i}C_{i}−1, respectively, the removed and trapped electron resides mainly on Si_{2C}. The energy of the charge neutral C-type C_{i}C_{i} defect is 0.11 eV higher than that of the A-type defect in the spin degenerate calculation, but 0.61 eV lower when spin polarization is taken into account. In addition, spin polarization lowers the energies of the +1 and −1 charge states by 0.20 eV and 0.23 eV, respectively.

Table III | Total energy differences (eV) for the ground states of the A- and B-C_{iCs} defects for different charges. The numbers in brackets refer to spin polarized calculations.

|             | A+−B+ | A−B0 | A−B− |
|-------------|-------|------|------|
| Experimenta | −0.02 | 0.02 | −0.04 |
| MINDO/3-Si_{38}b | −0.62 | 0.34 | ... |
| MINDO/3-Si_{41}b | −0.38 | 0.51 | ... |
| Clusterc | −0.43 | −0.35 | −0.50 |
| Local density approximationd | ... | 0.11 | ... |
| This work PBEsol | 0.15 [0.14] | 0.20 | 0.23 [0.23] |
| This work HSE | −0.09 [−0.29] | 0.04 | 0.07 [0.14] |

*aReference 37.
*bReference 38.
*cReference 19.
*dReference 20.

Figure 6 | Total DOSs of pristine Si (gray) and the A-, B-, and C-type C_{i}C_{i} defects (red) in the 0 charge state. The states left of the dotted line are occupied.
Figures 3(e), (f), and (g) present for different charge states the formation energy of the C₅Cs defect as a function of the Fermi energy. Note that for A-C₅Cs, B-C₅Cs, C-C₅Cs, and D-C₅Cs, the energies from the spin polarized calculations are used. The results for the A- and B-type configurations are similar, because of the tiny total energy difference between them. An exception is that A-C₅Cs has a higher (+/0) transition level. For C-C₅Cs, the charge states are favorable below 0.06 eV and above 0.91 eV, respectively. The Fermi energy range in between is dominated by the 0 charge state, as spin polarization of the two unpaired electrons on the C atoms lowers the total energy significantly.

Defect clusters. Two configurations of C₅O(SiI), denoted as C₅O(SiI)ₐ and C₅O(SiI)ₐ, see Figs. 1(c) and (d), are calculated. They are distinguished by the Si interstitial position relative to the C₅O pair and are similar to the C₄(a) and C₄(c) configurations reported by Backlund and Estreicher, respectively. In the figures the two Si atoms subject to a significant change in their position in the two configurations are marked in gray. The C and O interstitials form a ring with Si atoms. The four C-Si bond lengths are 1.84 Å, 1.85 Å, 1.89 Å, and 1.89 Å and the three O-Si bond lengths 1.77 Å, 1.79 Å, and 1.84 Å. The total energy of C₅O(SiI)ₐ is always 0.28 eV lower than that of C₅O(SiI)ₐ for the 0, 1, and 2 charge states. However, C₅O(SiI)ₐ turns out to be 0.45 eV and 1.51 eV higher in energy than C₅O(SiI)ₐ for the +1 and +2 charge state, respectively. The formation energy as a function of the Fermi energy for the two C₅O(SiI) defect clusters is plotted in Figs. 3(c) and (d). C₅O(SiI)ₐ has a lower formation energy than C₅O(SiI)ₐ, and a transition from the +1 into the 0 charge state is found for C₅O(SiI)ₐ as the Fermi energy increases. Transition levels between charge states are given in Table I. Still, C₅O(SiI)ₐ dominates in the whole Fermi energy range.

Total DOSs for the 0, 1, and 2 charge states of the two C₅O(SiI) configurations are shown in Fig. 7. In the structure of C₅O(SiI)ₐ, the Si₁-C bond is oriented almost perpendicular to the C-O ring with a bond length of 1.91 Å, which is slightly larger than the calculated Si-C bond distance of 1.89 Å in face centered cubic SiC. The distances between Siₐ and its Si neighbors are close to 2.50 Å (the shortest amounting to 2.44 Å), which is larger than the Si-Si bond length of 2.35 Å in the perfect lattice. This implies that Siₐ interacts little with the surrounding Si atoms so that it can be considered to be isolated. Partial DOSs (not shown) demonstrate that the states between 6 eV and 6.8 eV in the case of C₅O(SiI)ₐ are defect states largely resulting from the isolated Siₐ. We observe that the pseudo-gap between the valence band and the defect states increases in the +1 and +2 charge states. In the positively charged states the removed electrons stem mainly from Siₐ. The Si₁-C bond length reduces from 1.91 Å in C₅O(SiI)ₐ to 1.87 Å in C₅O(SiI)ₐ, which indicates that the interaction between Siₐ and C becomes stronger. The significant modifications of the electronic structure result in large energy gains for C₅O(SiI)ₐ and C₅O(SiI)ₐ.

Table IV | Binding energies (eV) and comparison to previous theoretical results

| Defect | This work | Literature |
|--------|-----------|------------|
| C₅(Si) | -2.34     | -2.0⁵      |
| C₅O   | -1.60     | -1.6⁴      |
| A-C₅Cₕ | -1.07     | -0.93⁵, -1.34⁴ |
| B-C₅Cₕ | -1.12     | -0.95⁵, -1.41⁴ |
| Si₁Cₕ | -1.68     | -1.28⁵, -1.76⁴ |
| C₅O(SiI)ₐ(CO₂+SiI) | -1.54 | -1.50⁶ |
| C₅O(SiI)ₐ(CO₂+SiI) | -1.26 | -1.11⁷ |
| C₅O(SiI)ₐ(CSi₂+O) | -0.81 | -1.57⁷ |
| C₅O(SiI)ₐ(CSi₂+O) | -0.53 | -1.18⁷ |

References:
⁵Reference 27.
⁶Reference 25.
⁷Reference 31.
⁸Reference 49.
Comparing the atomic configuration of $C_{\text{Si}}(Si)$ to that of $C_{\text{Si}}(Si)_{\text{n}}$, shows that the isolated Si forces another Si off its lattice site, with a (101) symmetry plane. In the electronic structure, the defect states shift to lower energy and merge to the valence band so that the donation of electrons is harder in $C_{\text{Si}}(Si)_{\text{n}}$. The stability of $C_{\text{Si}}(Si)$ and $C_{\text{Si}}(Si)_{\text{n}}$ in different charge states is determined by the competition between the energies associated with the defect states and the symmetry reduction. We observe that $C_{\text{Si}}(Si)_{\text{n}}$ is more stable than $C_{\text{Si}}(Si)$ in positively charged states, because electrons can be easier donated, which overcompensates the symmetry reduction. Nonetheless, $C_{\text{Si}}(Si)_{\text{n}}$ is more stable for other charge states due to its higher symmetry.

We find rather high values for the formation energies of the $C_{\text{Si}}(Si)$, $C_{\text{Si}}$, $C_{\text{Si}}C_{\text{Si}}$, and $C_{\text{Si}}(Si)_{\text{n}}$ defects. For example, in the 0 charge states of $C_{\text{Si}}(Si)$, $C_{\text{Si}}$, $A$, $B$, and $C$-type $C_{\text{Si}}$, $C_{\text{Si}}(Si)_{\text{n}}$, $C_{\text{Si}}(Si)$, and $C_{\text{Si}}(Si)_{\text{n}}$, they amount to 6.53 eV, 4.85 eV, 5.64 eV, 5.60 eV, 5.04 eV, 7.67 eV, and 7.95 eV, respectively. This is reasonable, because $C_{\text{Si}}(Si)$ and $C_{\text{Si}}Cs$ pairs form under irradiation. Furthermore, when the irradiation dose is increased more Si self-interstitials are created and trapped by C and $C_{\text{Si}}$, to form $C_{\text{Si}}(Si)$ and $C_{\text{Si}}(Si)_{\text{n}}$ clusters. This implies that the formation of $C_{\text{Si}}(Si)$ and $C_{\text{Si}}(Si)_{\text{n}}$ requires more energy, consistent with our results. Binding energies for defect pairs and clusters are reported in Table IV, which includes also previous theoretical results for comparison. Note that the total energy of $Si$ at the hexagonal site is taken to calculate the binding energies. While our results are largely converged carefully to guarantee an accuracy of 0.01 eV. The lattice constant of Si is

\[
\Delta H(p_{\text{Si}}, p_{\text{Si}}) = E(p_{\text{Si}}, p_{\text{Si}}) - E(p_{\text{Si}}, 0) + \sum n_{\text{Si}}(p_{\text{Si}}) + n_{\text{Si}}(0)
\]

where $E(p_{\text{Si}}, p_{\text{Si}})$ is the total energy of the defective cell with charge $q$, $E(p_{\text{Si}}, 0)$ is the total energy of the perfect Si cell, $n_{\text{Si}}$ represents the numbers of atoms added to or removed from the defective cell, and $p_{\text{Si}}$ corresponds to their chemical potentials. Moreover, $p_{\text{Si}}$ is the Fermi energy and is measured from the valence band maximum, having values inside the band gap, $E_{\text{VBM}} \leq p_{\text{Si}} \leq E_{\text{CBM}} + E_F$. The O and C chemical potentials are calculated using $\alpha$-quartz SiO$_2$, as (SiO$_{\text{2}}$)$_{\text{3}}$SiO$_{\text{3}}$, and face centered cubic SiC, respectively.

A way to investigate the energetics of point defect association is the calculation of the binding energies of the clusters. For example, the binding energy of a C$_{\text{Si}}$ atom to a C$_{\text{Si}}$ atom forms a C$_{\text{Si}}$ defect pair is given by

\[
E_b(C_{\text{Si}}C_{\text{Si}}Si_{\text{n}}) = E(C_{\text{Si}}C_{\text{Si}}Si_{\text{n}}) - E(C_{\text{Si}}) - E(C_{\text{Si}}Si_{\text{n}}) + E(Si_{\text{n}})
\]

where $E(C_{\text{Si}}C_{\text{Si}}Si_{\text{n}})$ is the energy of an N site supercell (here $N = 64$) with N–1 Si atoms, one C$_{\text{Si}}$ atom and one C$_{\text{Si}}$. $E(C_{\text{Si}})$ is the energy of a supercell containing one C$_{\text{Si}}$ atom and N–1 Si atoms, $E(Si_{\text{n}})$ is the energy of an N atom Si supercell. This definition a negative binding energy corresponds to a defect pair which is stable with respect to its constituent point defects.

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Author contributions
H.W. performed the calculations and analysed the results together with A.C. and U.S. C.A.L. and E.N.S. contributed to the discussion. All authors have proofread the manuscript.

Additional information
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