The Impact of Surface Size on the Radiative Thermal Behavior of Embedded Systems

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Abstract

This paper introduces a new global analytical model of the heat dissipation process that occurs in passively-cooled embedded systems, and explicit under what circumstances the folklore assumption that exponential cooling laws apply in such context is valid. Since the power consumption and reliability of microprocessors are highly dependent on temperature, both designers and, later on, run-time temperature management units must be able rely upon accurate heating and cooling models to handle heat generation and peak temperature. If exponential cooling models are justified for actively-cooled microprocessors, e.g., by forced air or water cooling, for passively cooled processors however, as frequently found in embedded systems such as mobile phones, an exponential law may not be theoretically justified. Here, we analyzed the tractability of the exact cooling law for a passively cooled body, subject to radiative cooling and a modest level of heat loss via convection. Focusing then on embedded microprocessors, we compare the performance difference between our new passive cooling law and the conventionally-used exponential one. We show that the differences between the exact solution and the exponential cooling law are not significant, and even negligible, for small surfaces of the order 10 cm\textsuperscript{2}. However, for larger surface sizes, the radiative cooling component may become comparable to the convective cooling component. Our results thus suggest that, in the absence of accurate temperature measurements, an exponential cooling law is accurate enough for small-sized SoC systems that require low processing overhead.

1 Introduction

With microprocessor manufacturing technologies keeping pushing the boundaries of nano-scale transistor dimensions, the amount of functionalities integrated on a single die increases accordingly. To afford such increased performance while ensuring the long-term viability of these systems though, proper temperature and power management become critical issues.

Excessive temperatures affect the reliability of circuit operations, but also inflate the energy consumption of microprocessors. Thermal gradients that occur both in space and time, induced by the variability in microprocessor load and operations, generate thermal cycles that have an adverse affect on the failure rate of systems \cite{1}. A 10\textdegree C to 15\textdegree C temperature increase may halve a microprocessor’s lifetime \cite{2}. The International Technology Roadmap for Semiconductors (ITRS) even states that processor costs and performance specifications may be limited by the lifetime reliability and is of primary concern in the microprocessor’s design phase \cite{3}. Since power consumption increases exponentially with increasing silicon temperature \cite{4}, thermal management techniques are necessary to avoid self-destruction, to increase the Mean Time To Failure (MTTF) and minimize power consumption. Such thermal techniques may be used at the design phase or can be deployed...
dynamically at run time by Thermal Management Units (TMUs) or Dynamic Thermal Management (DTM) systems. Multiple thermal control methods for microprocessors exist and show trade-offs between temperature profile, frequency settings, power consumption and implementation complexity.

Thermal control methods often incorporate a model describing the temporal thermal behavior of the microprocessor. Exponential-based laws are popular, and scientifically sound for systems without internal heat generation and subject to active cooling, e.g., forced air or water cooling. It has also been shown that an exponential law may be a good approximation when simulating the conduction of heat, under very specific circumstances: when the average system temperature is not too large and the system conducts heat much faster than it gains heat from its surroundings. However, passively cooled systems, as frequently found in embedded devices such as mobile phones or Systems-on-Chip (SoC), are not forcibly cooled. These are subject to the same physical laws for dissipating their heat to the environment, but rely on different aspects of the heat dissipation process.

In this paper, we develop an exact analytical solution to the problem of modeling the passive cooling of embedded systems. It is important to understand the difference between an exponential cooling law and the cooling law of passively cooled devices since, in the literature, the radiative cooling aspect is frequently neglected. We believe that this is because of its non-linear behavior, which introduces problems in mathematical formulations. In the case of active cooling, convective heat transfer dominates the other heat transfer modes whereas, for passive cooling, radiation may become equally important, sometimes even more important, and may dominate the convective heat transfer mode. When radiation cannot be neglected, the temporal thermal behavior of the system will deviate from an exponential cooling law. In this paper we analyze under which circumstances the radiation may be significant enough for it not to be neglected. We show that the size of the microprocessor plays an important role in this question. However, in cases where the cooling surface of the microprocessor is small (<10 cm), the difference between an exponential cooling law and the passive cooling law is not significant. Based on the passive cooling law’s complex formulation and in the absence of accurate temperature measurement samples, our work therefore suggests that an exponential cooling law is accurate enough for SoC applications and systems that require low processing overhead.

The main contributions of this paper are:

- the exact analytical solution for the problem of (passive) cooling of a body subject to radiation, convection, and internal heat conversion;
- approximations to the exact analytical solution for use in practical applications;
- actionable rules-of-thumb to assess when passive cooling becomes non-negligible compared to active cooling in embedded systems;
- prototype implementations of the exact and approximate solutions for passively cooled objects (see Appendix D).

The rest of the document is developed as follows. Section 2 highlights the use of cooling laws in existing research related to thermal management units. Section 3 develops the exact cooling law for microprocessors subject to passive cooling; this law is also validated via finite-element simulations and approximations are analyzed. Besides, the impact of active cooling of microprocessors is also discussed. Section 4 studies the performance difference between the exponential cooling law and the passive cooling law. We conclude in Section 5 with a summary and give directions for future research.

## 2 Cooling in Thermal Management Techniques

Thermal management techniques for microprocessors have been devised to limit the heat dissipation of microprocessors. Excessive heat dissipation may have adverse effects on performance and the short term and long term failure rate of the microprocessor. Basic run-time thermal management techniques can be rudimentary, such as clock gating. Yet, if service continuation is needed, more
advanced thermal techniques are required. Thermal-aware design of microprocessors can also be effective to minimize heat dissipation during run time. The challenge here, however, lies in decision making based on incomplete design details.

To get a current perspective on how such issues are addressed in the literature, we surveyed top computer architecture and Very-Large-Scale Integration (VLSI) conferences for papers devoted to TMUs, DTM and temperature-aware design methods based on heat transfer theory. The conferences surveyed are ISCA, MICRO, ASPLOS, HPCA, PACT, ISLPED, ICCAD, DAC, DATE, ASP-DAC from 2010 to 2014. We identified 35 papers focusing on the thermal optimization of microprocessors using heat transfer models. 90% of these papers base their results solely on simulation or numerical analysis; the remaining ones use either pure measurements or a combination of simulation and measurements to make their point. Beside custom thermal simulators and models, non-commercial and open-source thermal simulators are mostly used: these are based on finite-element methodologies. Commercial applications such as COMSOL Multiphysics®, Autodesk Simulation CFD or FLoTHERM®, which support the radiative heat transfer mode, are not used in the selected papers. About 40% of the selected papers deploy Hotspot for their thermal simulations. Hotspot [7] is a self-proclaimed accurate and fast thermal model designed for microprocessor architectural analysis, e.g., floor planning. The basic setup of Hotspot includes active cooling via a heat sink. No passive cooling possibilities are available in Hotspot. Other experimental simulators, such as LightSim [8], CONTILTS [9], ISAC [10] and PowerBlurr [11], also allow for thermal analysis of microprocessors, but are less popular and none support radiative cooling. In most of the simulations, the temperature at steady state and transient temperatures are available, where the steady-state case is much faster to compute than the transient behavior one.

It is worthwhile to wonder why no non-commercial simulators support radiative cooling. One reason could be that the non-linear behavior of radiation is not easy to handle in mathematical formulations, or that maybe it is not clear to what extent radiation actually affects the thermal behavior of semiconductors. Given the lack of handling of passive cooling in most simulators, it is not surprising that passive cooling has not gotten much attention in the thermal management research community. In fact, we only found one paper [12], about 3D integrated circuits, that mentions that radiation may influence the thermal behavior of microprocessors; yet in this work no further reference to radiation is found. And yet, 30% of the papers we surveyed claim that their research is applicable to embedded systems, a situation in which passive cooling is usually of the essence.

Beside generic thermal microprocessor simulators, dedicated embedded system thermal simulators were also developed. Therminator [13], for example, is a thermal simulator designed to simulate heat dissipation in smartphones. Finite element methodologies are used to compute the heat propagation through an arbitrary smartphone configuration, which may include a screen printed circuit board (PCB), battery, case, display etc. The authors show that their dedicated thermal simulator produces results that are close to what commercial computational fluid dynamics (CFD) software would calculate. Therminator takes the convective and conduction heat transfer modes into account. Heat loss via radiation, however, is not implemented in their thermal simulator. Luo et al. [14] analyzed the issue of thermal management on mobile phones based on numerical simulation and basic thermal models. The authors came up with design proposals on how to improve the thermal management of mobile phones by studying the steady-state behavior of the system. Even though radiation is mentioned in the introduction including formulations, radiation is not present in their steady-state analysis. Gurrum et al. [15] decomposed, just as Luo et al. [14], a hand-held device in multiple subparts with different physical properties and analyzed its thermal behavior. Radiation, however, did not come to their attention.

From our literature survey we conclude that the numerical tools used for thermal behavior of embedded systems can be classified into three categories. First, we have the general-purpose CFD software, which is able to simulate arbitrary systems including all modes of heat transfer. These systems require the most efforts to produce interesting results. The second class corresponds to dedicated embedded system simulators. We have observed that the designers of the simulators are aware of surface radiation but they do not provide support in their simulators. And last, which are the most popular, are the generic microprocessor thermal simulators. We have not seen any of these microprocessor simulators supporting the radiative heat transfer mode. This provided us with a strong motivation for our work, which strives to understand the possible impact of radiation
on the transient and steady-state thermal behaviors of micro-processors in the context of embedded systems.

3 Cooling Laws for Microprocessors

The exponential cooling law is the most widely used cooling law to model the thermal behavior of microprocessors. The rationale behind an exponential law is based on temperature traces of forcibly cooled microprocessors, which indeed show clear exponential behavior \[14, 15\]. One may attribute the exponential curve to Newton’s law of cooling. However, the presence of internal heat conversion, which renders the applicability of Newton’s law of cooling irrelevant for microprocessors, should not be forgotten. In the sequel we show however that Newton’s law of cooling extended with internal heat conversion also yields an exponential cooling law. For passively cooled microprocessors, the radiative heat transfer mode, beside natural convection, also needs to be taken into account.

In this section, after a brief overview of basic heat transfer principles, we develop the cooling law for an actively cooled body. We then extend this model with radiative cooling to obtain our first contribution, a representative model for passively cooled systems.

3.1 Basics of Heat Transfer

Heat transfer happens via a combination of the three fundamental modes: convection, conduction, and radiation. Each of these modes follows its respective law. In the sequel we assume an isothermal body that cools via convection and radiation. Isothermal conditions may be approximated if the body heats up uniformly, or if the internal heat conduction happens considerably faster than the heat loss by the body to the environment. Therefore we don’t discuss conduction in detail.

A solid body exposed to a moving fluid is subject to energy exchange if the temperatures of the body and the environment differ. Energy is *convected* from or to the body if the moving fluid has a different temperature from the body. The energy transfer rate \( q \) between the moving fluid and the surface of the body is formally known as Newton’s law of cooling:

\[
q = \frac{dT}{dt} = hS(T_s - T_a),
\]

where \( T_a \) is the temperature of the moving fluid (environment), \( T_s \), the temperature of the body’s surface, and \( h \), the *convective heat transfer coefficient* [W/(m\(^2\)·K)].

Radiative heat transfer happens through exchange of electromagnetic waves, possible through both vacuum and transparent media. Stefan-Boltzmann’s law states that the power radiated from a *blackbody* is proportional to its temperature. A *blackbody* is a body that absorbs all incident radiation. In particular, Stefan-Boltzmann’s law states that the radiative heat transfer \( q \) [W/m\(^2\)] is proportional to the blackbody’s temperature to the 4th power:

\[
q = \varepsilon \sigma S T_4,
\]

where \( \sigma \) is the Boltzmann constant \( 5.6697 \cdot 10^{-8} \) [W/(m\(^2\)·K\(^4\)]], \( S \), the size of the surface, and \( \varepsilon \in [0, 1] \), the *emissivity* of a gray body’s surface (dimensionless). A *gray body* is a body that reflects a certain amount of the incident radiation. The emission and absorption of a gray body can be well represented by a blackbody’s behavior scaled by its emissivity: \( \varepsilon < 1 \). In practical situations the total heat loss of a body via radiation is equal the emitted radiation minus the absorbed radiation:

\[
q = \varepsilon \sigma S(T_4 - T_a^4),
\]

where \( T_a \) is the radiation temperature of the environment. Here we implicitly assumed that the environment has the same emissivity as the body itself.

The total heat transfer from a body happens via the combination of the basic heat transfer modes. Besides, a body may also produce heat \( H(\cdot) \) [W] which is referred to as *internal heat conversion*. The internal heat conversion is a function of space, time, temperature or others. In the sequel we will assume that the internal heat conversion is homogeneously present in the entire body.
3.2 Active Cooling of Microprocessors: the Newtonian Approach

Newton’s law of cooling states that the temperature rate of change of an object is proportional to the difference between the ambient temperature and the object’s temperature. Thermal Management Units (TMUs) and Dynamic Thermal Managements (DTMs) often assume the system to cool down following Newton’s law of cooling. Accordingly it is implicitly assumed that the system’s power consumption as a result also exhibits exponential behavior over time, comparable to an RC network. Such assumptions are frequently found in thermal management systems [1, 16, 17, 18, 19, 20, 21]. For actively cooled systems, i.e., cooled using forced convection (such as computer fans or water cooling), an exponential assumption is a good approximation when radiative and conductive cooling may be neglected, as we explain now.

Assume that for such systems the stored energy is approximated by the sum of the heat transfer induced by convective cooling: \( h(T_a - T) \), and an internal heat conversion, which we deem linear as a first-order approximation \((\eta_1 T + \eta_0)\):

\[
\frac{dT}{dt} = C - \text{convection} + \text{internal heat conversion}
\]

\[
= hS(T_a - T) + (\eta_1 T + \eta_0).
\]

(4)

where \( C \) is the system’s heat capacity. Note that, if the active cooling system consists of a fan and heat sink, then \( h \) depends upon the dimensions of the heat sink or Central Processing Unit (CPU) surface area, and the revolutions per minute (rpm) of the fan. Moreover, \( \eta_1 \) and \( \eta_0 \) are also dependent on the operating frequency of the microprocessor and the load on the system. Similar to Weissel and Bellosa’s [22] work, one gets, from Equation 4:

\[
\frac{dT}{dt} = hS(T_a - T) + (\eta_1 T + \eta_0)
\]

\[
\frac{1}{hS - \eta_1} \int (T - T_a) \, dt = -(hS - \eta_1)T + (\eta_0 + hST_a)
\]

\[
\ln \left( \frac{T - \eta_0 + hST_a}{hS - \eta_1} \right) = \frac{(hS - \eta_1)}{C} \int T - \eta_0 + hST_a \, dt + c_0
\]

\[
T - \eta_0 + hST_a = c_0 e^{-\frac{(hS - \eta_1)}{C} t},
\]

(5)

while imposing the initial condition at \( t = 0 \): \( T(0) = T_0 \). Therefore \( c_0 = T_0 - \frac{\eta_0 + hST_a}{hS - \eta_1} \), and thence

\[
T_{ac}(t) = \frac{\eta_0 + hST_a}{hS - \eta_1} + \left( T_0 - \frac{\eta_0 + hST_a}{hS - \eta_1} \right) e^{-\frac{(hS - \eta_1)}{C} t}.
\]

(6)

It is clear that such a system is only stable if the cooling process with constant \( h \) convects heat away from the system faster than the system is generating internal heat. The system is stable if there exists an equilibrium temperature \( T_e \) for the system, which is equivalent to saying that

\[
0 = hS(T_a - T_e) + (\eta_1 T_e + \eta_0) \Rightarrow h = \frac{\eta_1 T_e + \eta_0}{S(T_e - T_a)}.
\]

(7)

where all constants \( \{h, T_a, T_e, \eta_1, \eta_0\} \in \mathbb{R}^+ \). We can state, given that \( h \) must be positive, that \( T_e > T_a \). We can also conclude from Equation 6 that \( h \) is always larger than \( \eta_1/S \). If \( h < \eta_1/S \), the exponent in Equation 6 would go to infinity. In practical applications the value of \( h \) must be dimensioned properly such that the system’s \( T_e \) stays below the maximum operation temperature.

Not surprisingly, Newtonian cooling with linear internal heat conversion yields again an exponential relationship between temperature and time. Consequently, the power \( P \) consumed by the system, which is an affine transformation of temperature \( P = \eta_1 T + \eta_0 \), will also exhibit exponential behavior. An exponential model for actively cooled systems with linear (or constant, \( \eta_1 = 0 \)) internal heat conversion is therefore a valid approximation. The exponential assumption
is however not quite the same as assuming simple Newtonian cooling, as the coefficients in both models are different, mainly due to the presence of the internal heat conversion. In the case of the presence of internal heat conversion the equilibrium temperature \( T_e \) of the system will be larger than the ambient temperature \( T_a \):

\[
T_e = T(\infty) = \frac{\eta_0 + hST_a}{hS - \eta_1} + \left( T_0 - \frac{\eta_0 + hST_a}{hS - \eta_1} \right) e^{-\infty} = \frac{\eta_0 + hST_a}{hS - \eta_1}.
\]

This is the same result as obtained in Equation 7.

If the internal heat conversion follows a second-order polynomial, similar results can be derived. The interested reader can check that the solution is in this case is given by:

\[
Cd\frac{dT}{dt} = hS(T_a - T) + (\eta_2T^2 + \eta_1T + \eta_0) \quad T_{ac}(t) = \frac{\omega_1 + \omega_2e^{-\frac{k_2}{C}t}}{1 + e^{-\frac{k_2}{C}t}} + c_o,
\]

where \( k_2, c_o \) and \( \omega_1 \) are constants derived from the roots of the expression on the right-hand side of the first formula.

### 3.3 Passive Cooling under Radiation, (Natural) Convection and Internal Heat Conversion

We now extend the previous model to better fit passively cooled embedded systems. Microprocessors that are not actively cooled must indeed rely on passive cooling to attain a temperature equilibrium state. Passive cooling mechanisms include radiation, and also convection. Note though that convection, in this case, may be considerably smaller than when the system is actively cooled. The convection arising here may be originating from buoyancy forces, or natural movement of air, e.g., wind. For this reason, in this particular case, sometimes the convection is referred to as **natural convection** as the movement of air is not enforced on the system.

Assume an isothermal body subject to radiative cooling and convection with internal heat conversion. The temperature change of such an object at any given point in time is equal to the heat absorbed from the environment, plus the internal heat conversion, minus the heat released to the environment. Absorption of heat happens via radiation whereas the release of heat is happening both via radiation and convection. The temperature change of such a system, with internal heat conversion, can be represented by the following equation:

\[
C\frac{dT}{dt} = \epsilon\sigma S(T_a^4 - T^4) + hS(T_a - T) + (\eta_1T + \eta_0), \quad (8)
\]

where \( C \) is the thermal capacity of the body, \( S \) is the radiation surface of the object, \( \epsilon \) is the emissivity of the body, and \( \sigma \) is the Boltzmann constant. Here it is assumed that the internal heat conversion is linearly dependent on the temperature of the body: \( H(T) = \eta_1T + \eta_0 \). Yet, higher order polynomials (up to the 4th order) can be used as well for the following derivation to hold (as we show in Appendix A).

By rearranging Equation 8 we obtain:

\[
\frac{dT}{dt} = \frac{1}{C}(-\epsilon\sigma ST^4 + (\eta_1 - hS)T + (\eta_0 + S(hT_a + \epsilon\sigma T_a^4))), \quad (9)
\]

Here, the right-hand side is a 4th-order polynomial. This becomes clear when the equation is rearranged again and the following variables are introduced: \( \kappa_4 = \frac{\epsilon\sigma}{C}, \kappa_1 = \frac{\eta_1 - hS}{C}, \) and \( \kappa_0 = \frac{\eta_0 + S(hT_a + \epsilon\sigma T_a^4)}{C} \). One then gets:

\[
\frac{dT}{dt} = -\kappa_4T^4 + \kappa_1T + \kappa_0. \quad (10)
\]
Table 1: Variables used for the COMSOL simulations. Specific values were calculated for the convective heat transfer coefficient (h) and internal heat conversion (ihc) such that a predefined equilibrium temperature is attained.

The derivation provided in Appendix B shows that the exact solution to the problem of cooling of a body subject to radiation, convection, and internal heat conversion is given by

\[
t = -\frac{1}{\kappa_4} \left( A \ln |T - \omega_1| + B \ln |T - \omega_2| + \frac{C}{2} \ln |(T - \alpha)^2 + \beta^2| + \frac{\alpha C - D}{\beta} \arctan \left( \frac{T - \alpha}{\beta} \right) + c_o \right),
\]

where \( c_o \) must satisfy the initial conditions \( f(T_0) = 0 \), if \( f(T) \) denotes the right-hand side expression in Formula (11):

\[
c_o = -A \ln |T_0 - \omega_1| - B \ln |T_0 - \omega_2| - \frac{C}{2} \ln |(T_0 - \alpha)^2 + \beta^2| - \frac{\alpha C + D}{\beta} \arctan \left( \frac{T_0 - \alpha}{\beta} \right),
\]

the \( \omega_i \) are the roots of the 4th-order polynomial given in Equation (10) and

\[
A = \frac{1}{(\omega_1 - \omega_2)((\Re(\omega_3)^2 + \Im(\omega_3)^2) - \omega_1(2\Re(\omega_3) - \omega_1))}
\]

\[
B = -A \frac{\Re(\omega_1)^2 + \Im(\omega_1)^2 - \omega_1(2\Re(\omega_1) - \omega_1)}{\Re(\omega_3)^2 + \Im(\omega_3)^2 - \omega_2(2\Re(\omega_3) - \omega_2)}
\]

\[
C = - (A + B)
\]

\[
D = A(2\Re(\omega_3) - \omega_1) + B(2\Re(\omega_1) - \omega_2)
\]

where \( \Re \) and \( \Im \) denote the real and imaginary parts of complex numbers.

Similarly to actively cooled system, the passively cooled system will tend towards an equilibrium temperature \( T_e \) only if Equation (11) equates to zero. Given that an equilibrium temperature exists, the convection heat transfer coefficient \( h \) must be such that

\[
0 = (\eta_1 - hS)T_e - \epsilon \sigma S T_e^4 + (\eta_0 + hS T_a + \epsilon \sigma S T_a^4) \Rightarrow h = \frac{\eta_1 T_e + \eta_0 + \epsilon \sigma S (T_e^4 - T_a^4)}{S(T_e - T_a)},
\]

where all constants \{\( h, T_e, T_a, \epsilon, S, \eta_1, \eta_0 \)\} ∈ \( \mathbb{R}^+ \). Consequently, this is only possible if \( T_e > T_a \), as in the case of active cooling.

The exact solution for passively cooled objects as presented in Equation (11) is a function of the temperature: \( t(T) \). For practical reasons, such as for proportional-integral-derivative (PID) control techniques, an analytical formulation in the form of \( T(t) \) would be preferred. Inverting the exact solution is however, not a straightforward task, mainly because the \( \arctan \) is hard to deal with as it keeps recurring. Numerical approaches will thus be preferred to compute the exact solution. In Section 3.5 we will discuss approximations to the exact solution.

3.4 Experimental Validation of the Exact Cooling Law

To validate the passive cooling solution defined in Equation (11), we setup a set of CFD simulations in COMSOL where we analyze the transient thermal behavior of a slice of silica glass. A 3D conjugate heat transfer scenario was created, with simulation settings as shown in Table 1. The exact same values, as listed in this table, were also used in our theoretical model. To simulate an isothermal
object in COMSOL we have multiplied the thermal conductivity of the silica glass by $10^3$. The silica glass has a surface area of 0.1 m$^2$. For the heating process $T_0$ is set to 25°C and $T_e$ is scaled between $T_0$ and 45°C. Similarly, for the cooling process $T_0=45$°C and $T_e$ is scaled between $T_0$ and 25°C. We used linear internal heat conversion with the parameters as shown in Table 1. The convective heat transfer coefficient was set (Equation 13) such that with the given internal heat conversion the predefined equilibrium temperature is attained.

Figure 4 shows the transient thermal behavior of the silica glass as described above. On the left the case for maximum internal heat conversion is shown and on the right for minimum internal heat conversion. Both the cooling and heating process are shown in the same graph. We have generated data for various surface sizes and equilibrium temperatures; since all graphs look similar we don’t show all of them. Our theoretical model curves follow the experimental COMSOL curves well. The maximum temperature difference between our model and the COMSOL results is less than 0.5°C. Interestingly, the COMSOL transient data seems to have a slightly steeper slope than our theoretical model. This could be originating from the fact that the COMSOL object is not 100% isothermal. Despite the small temperature discrepancy between our analytical model and the COMSOL data we may deem our model an appropriate solution for passive cooling with internal heat conversion.

3.5 Approximations of the Passive Cooling Law $f(T) = t$

The exact solution for the passive heat Equation 11 is of the form $f(T) = t$. Ideally, for practical motivations, we would like to know the inverse $f(t) = T$. For example, this may be convenient for the equation to be used in PID controller systems. Calculating the inverse of Equation 11 is, however, a challenging endeavor. Therefore, we will assume some simplifications to obtain an invertible heat equation.

Finding a useful expression $f(t) = T$ requires isolating $T$ in Equation 11. Mainly the presence of the arctan throws a monkey wrench in the mathematical derivation. Linearization or differential approximation will not provide any help as the derivative within the pertinent temperature range, i.e. between 25°C and 55°C, is far from being constant. Converting the arctan into a logarithm introduces imaginary numbers; yet, applying complex exponentiation rules will not get rid of the arctan. The arctan keeps recurring further on in the derivation. So we need to walk different paths to come to a solution for $f(t) = T$.

Table 2 shows an overview of three different approximations that we will consider. The derivation and motivation behind each approximation, as well as the definition of all the variables, are expounded in Appendix C. In short, the coefficient approximation models the radiation within a specific temperature range with a quadratic polynomial. This reduces Equation 11 to a second-order problem. The first and second O’Sullivan approximations are based on a binomial expansion.
that mingles the coefficients of Equation [10] in a deterministic manner. The advantage is that the resulting equation is invertible when higher-order coefficients are dropped. Also, the accuracy of the approximations can be controlled by the degree of coefficients selected. As can be observed from Table 2, the coefficient approximation and the second-order O’Sullivan approximation are similar in shape. However, the definition of their respective variables have no common ground.

Let us analyze the accuracy of the approximations. We define the measure of accuracy as the root-mean-square error (RMSE) between the exact cooling solution \( \phi \) and an approximate solution \( \psi \) for \( n \) samples:

\[
RMSE = \sqrt{\frac{\sum_{i=0}^{n}(\phi_i - \psi_i)^2}{n}},
\]

where \( n \) is the number of samples over which RMSE is computed. We defined \( n=500 \) and equally spaced between \( t \in \{0, f(0.99 \cdot T_o)\} \) (see Equation [11] for \( f(T)=t \)). The exact cooling law and its approximations are generated with the same constants as the COMSOL simulation in the previous section; these are listed in Table 1. We investigate the accuracy while changing surface area \( S \), internal heat conversion (ihc), equilibrium temperature \( T_e \), and for the cooling and warming process separately. We set \( T_0=25^\circ C \) for the heating process and \( T_0=55^\circ C \) for the cooling process. We vary the equilibrium temperature \( T_e \) between 25°C and 55°C. The convective heat transfer coefficient is computed accordingly to attain the respective equilibrium temperature based on Equation [15]. The variables generated for the exact cooling law are then used to compute the approximations.

Figure 2 shows the RMSE of the approximations for different surface areas, internal heat conversion and equilibrium temperature settings. From all graphs it is clear that the coefficient approximation is performing best. Also, the second-order O’Sullivan approximation is considerably better than the first-order O’Sullivan approximation. However, for very small surface sizes the errors in all approximations are acceptable. Interestingly, the first-order O’Sullivan approximation does well for small surface sizes, because the radiative part in the heat equation becomes negligible for smaller surface sizes, and so the passive heat equations tends towards an exponential cooling law. Consequently the first-order O’Sullivan approximation, being an exponential function, is able to approximate the exact cooling law well for very small surface sizes.

The errors for small internal heat conversion seem to be systematically larger than the errors for the maximum internal heat conversion case. The same observation can be made for the heating and cooling processes. The heating process seems to be more erroneous than the cooling process.

For variable equilibrium temperatures we see that for \( |T_0-T_e| \) the error increases for the heating process and decreases for the cooling process. In the derivation of the O’Sullivan approximations we have assumed that \( T-T_e \) remains relatively small. This implies that the larger \( T \) departs from \( T_o \) the more imprecise the approximation becomes. \( T_o \) was set to 20°C for the cooling process \( T_0=55^\circ C \) and the equilibrium temperature \( T_e \) was scaled between 25°C and 55°C. Similarly for the heating process \( T_0 \) was set to 55°C and \( T_e \) was scaled between 25°C and 55°C. Thus as the cooling process approaches \( T_o \) for increasing \( |T_0-T_e| \), \( T-T_e \) becomes smaller, and hence also the error between the O’Sullivan approximations and the exact cooling law. The reverse observation is also valid for the heating process; RMSE becomes larger for larger values of \( T-T_o \). The error properties in the case of the coefficient approximation is dependent on the fit of the second-order polynomial on the (quadratic) radiation function.

Overall, we do not advise to use the first-order O’Sullivan approximation, unless the surface area is really small. The second-order O’Sullivan approximation can be used but with caution. The equilibrium temperature should not depart too much from the ambient temperature \( T_o \); \( T -
Figure 2: Root mean-square error between the exact cooling law and the approximations. On top (a,b) the surface area of the object under study is a variable ($T_e=55^\circ C$), whereas the equilibrium temperature $T_e$ is variable in the bottom graphs ($S=0.01m^2$) (c,d). The Coefficient Approximation seems to outperform the other approximations. The Second-order O’Sullivan approximation is performing acceptably as well for small values of $\|T−T_a\|$.
$T_a < 15^\circ C$ seems acceptable. We recommend, however, the use of the coefficient approximation, even though the solution isn’t much elegant when other large polynomial coefficients are introduced.

### 4 Comparing the Passive and Active Cooling Laws

Given the intrinsic complexity of the (inverse) function describing passive cooling compared to the rather straightforward exponential specification of other cooling modes, it is worth investigating in which cases dealing with it is necessary in practice. We ran a series of simulations to understand under what circumstances the passive and active cooling laws differ from each other. The main difference between the active and the passive cooling law is the presence of the radiative heat transfer mode. Thus, if the radiative heat transfer is negligible compared to the convective heat transfer, the passive cooling law will approach an exponential cooling law. We explore when such situations occur based on concrete microprocessor use cases.

Let us recall the equations governing the cooling process of a passively and actively cooled isothermal body with internal heat conversion:

**Active cooling:**

$$C\frac{dT}{dt} = hS(T_a - T) + H(T)$$

**Passive cooling:**

$$C\frac{dT}{dt} = \epsilon \sigma S(T_a^4 - T^4) + hS(T_a - T) + H(T),$$

and their respective convective heat transfer coefficient $h$, at a given equilibrium temperature $T_e$:

**Active cooling:**

$$h_{ac} = \frac{H(T_e)}{S(T_e - T_a)}$$

**Passive cooling:**

$$h_{pc} = \frac{\epsilon \sigma S(T_a^4 - T_e^4) + H(T_e)}{S(T_e - T_a)},$$

where $H(T)$ is the function of the temperature $T$ defining the internal heat conversion of the body. We have shown previously that $H(T)$ is well described by an exponential equation \[4\]. Even more, within the temperature range $25^\circ C < T < 55^\circ C$, the exponential can be approximated well with a linear or quadratic polynomial. Yet, for the more extended temperature range $25^\circ C < T < 85^\circ C$, an exponential function is advised.

We compare the active and passive cooling of a microprocessor in the context of an embedded system, i.e., a low-power processor subject to internal heating conversion and cooling. In order to do so, we assume a simplified microprocessor model: an isothermal volume, with the physical properties of silica glass (SiO$_2$), and cooled via convection and radiation. Table 3 shows the values used in our simulations. The left column lists the fixed variables: $\sigma$, $\epsilon$, $T_a$, and $D$. We chose the emissivity of PVC for $\epsilon$ and fixed $T_a$ to be a representative room temperature. The height of our microprocessor $D$ is characteristic for a modern SoC. The other column lists the variables that may vary during the analysis. We study the impact of the surface size $S$ over which the device cools via convection and radiation. The minimum size was set to a square with a side of 1 cm. This is representative for a small SoC; for example, the Samsung Exynos 5 SoC has a side length of 1.6 cm. The maximum surface area was set to 0.1 m$^2$, which is a representative area for a large tablet. We
Figure 3: Ratio between the convective heat transfer coefficients of active and passive coolings at a given equilibrium temperatures $T_e$ (see curve labels in °C). In Figure (a) the internal heat conversion is set to a maximum, while in (b) it is set to a minimum. The vertical dashed lines represent typical surfaces of a SoC ($\approx 2.5 \text{ cm}^2$), a smartphone ($\approx 70 \text{ cm}^2$) and a tablet ($\approx 5 \text{ dm}^2$). A horizontal reference line is drawn at 0.95.

analyze the behavior of the system within the temperature range $T \in [25, 85]^{\circ} \text{C}$. Throughout the analysis, we define the internal heat conversion $H(T)$ to be an exponential function $\left(\alpha + e^{(T-\gamma)/\beta}\right)$; the coefficients are shown Table 3. The left values are for minimal internal heat conversion, the right values for maximum internal heat conversion. The values for $\alpha$, $\beta$ and $\gamma$ were derived from power and temperature measurements on a SoC sporting a CORTEX A15 [4]. We look at the system’s behavior when the A15 is running at full capacity, i.e. at 1.6 GHz, and when the A15 is running in low-power mode, i.e. at 800 MHz. The heat capacity of the body $C$ is the product of its volume $SD$ and its specific heat and density ($\approx 2.4 \times 10^6 \cdot 0.712$).

4.1 Relative Heat Transfers

First, we look at the ratio of the convective heat transfer coefficients of the passive and active cooling cases. The temperature $T_0$ at $t=0$ is set to 25°C, which leads to a series of equilibrium temperatures $T_e$. Then we compute the respective convective heat transfer coefficients as per Equation 16. The ratio of the convective heat transfer coefficients $r_{cr}$ is given by

$$r_{cr} = \frac{h_{pc}}{h_{ac}} = \frac{\sigma S(T_a^4 - T_e^4) + H(T_a)}{H(T_e)}$$

$r_{cr}$ shows how much the active and passive cooling laws will resemble. If $r_{cr} = 1$, there is no difference between the two cooling cases. The more $r_{cr}$ tends to zero, the more the two cooling laws will deviate in behavior.

Figure 3 shows the ratio of the convective heat transfer coefficient of the passive and active cooling cases. Given that $r_{cr}$ stays well above 0.95, it is observed that, for a small object, similar to SoCs (left most vertical dashed line), the difference between active and passive cooling will be very small for all equilibrium temperatures ranging between 20°C to 85°C. For a moderate surface size, e.g., the size of an average smartphone (middle vertical dashed line), the radiative cooling starts to become more prominent already for temperatures close to the ambient temperature $T_a$. For equilibrium temperatures more than about 5°C above $T_a$, signs of deviating behavior will become clearly visible. Large surface sizes and equilibrium temperatures close to $T_a$ will yield a $r_{cr}$ that is smaller than 0.95. This implies that the radiative cooling for large surfaces is most likely non negligible. As a general rule of thumb, we can say that the larger the equilibrium temperature and the cooling surface, the more behavioral differences between passive and active coolings will occur. So how large are the differences temperature-wise in particular?
4.2 Temperature Differences

When looking at the temperature differences between the passive and active coolings at specific points in time, we must differentiate between the cooling and heating processes. Convective heat transfer is proportional to the difference of the body temperature and the ambient temperature, and is therefore independent on the absolute temperature of the body and environment. This results in a symmetry between the heating and the cooling processes for convective heat transfer. The radiative heat transfer, on the other hand, is dependent on the absolute values of the body and the environment. This is illustrated as follows for the convective and radiative heat transfers respectively:

\[ |hS(T - (T - x))| = |hS(T - (T + x))| \]
\[ |\varepsilon\sigma S^4(T - (T - x))| \neq |\varepsilon\sigma S^4(T - (T + x))| \]  \hspace{1cm} (17)

As a consequence, due to the last inequality, the radiative heat transfer process will not be symmetric for the cooling and heating processes. Moreover, when radiative heat transfer is combined with convective heat transfer, the symmetry property of the heating and cooling processes will not hold either.

Let us define the temperature lag \( \Delta T \) between an actively and passively cooled identical body, measured at the moment when the passively cooled body reaches a reference temperature \( T_r \). The reference temperature \( T_r \) is henceforth defined as \( T_r = 0.85(T_e - T_0) + T_0 \), i.e., when the body temperature has reached 85% of its equilibrium temperature, starting from \( T_0 \). It is also assumed that both the passively and actively cooled bodies have the same internal heat conversion process and boundary condition \( T_0 \) at \( t = 0 \). Figure 4 shows the relative temperature lag \( \Delta \tau \), which is defined as the absolute temperature lag \( \Delta T \) divided by the temperature difference at the start and at equilibrium \( |T_e - T_0| \):

\[ \Delta \tau = \frac{\Delta T}{|T_e - T_0|} = \frac{T_{pe} - T_{we}}{|T_e - T_0|} \]  \hspace{1cm} (18)

The relative temperature lag \( \Delta \tau \) is depicted in Figure 4 for both a large and a small internal heat conversion, as defined before, and for the heating and cooling processes separately. A reference line is drawn for \( \Delta \tau = 5\% \). Data points on the right of the dashed blue line show configurations with one or more negative convective heat transfer coefficients. This implies that in these cases additional heat needs to be added to attain the given equilibrium temperature. These data points are however, not of concern in our work.

For the case of large internal heat conversion, the relative temperature lag \( \Delta \tau \) for small surfaces stays below 0.5%, meaning that the presence of radiative heating will be quasi unnoticeable here. \( \Delta \tau \) stays around 5% in the case of small internal heat conversion, which may be difficult to spot. Contemporary on-die processor temperature sensors report frequently temperatures in steps of 1°C. Given this quantization noise, a relative temperature lag of 5% could be hard to identify when \( |T_e - T_0| > 20^\circ C \). So for small processor temperature variations, it is again unlikely that a contemporary processor temperature sensor is able to distinguish between active and passive cooling. For a smartphone-size cooling surface, the relative temperature lag varies significantly depending on the situation. For a large internal heat conversion and heating, there is less than 5% difference between passive and active cooling. For the other cases, however, the discrepancy between the passive and active cooling can run up from nil to as high as 10%, depending on the equilibrium temperature. \( \Delta \tau = 10\% \) is already noticeable at \( |T_e - T_0| > 10^\circ C \) in the presence of 1°C quantization noise. The data for the tablet-sized cooling surfaces is mostly outside the valid surface area where the convective heat transfer coefficients for passive and active cooling are negative.

Generally speaking, we notice that the relative temperature lag \( \Delta \tau \) for the heating case is smaller than for the cooling case. This can be explained via the inequality of Equation (17). The radiative heat transfer coefficient will have greater weight when the body’s temperature is larger than the equilibrium temperature than when the temperature is below the equilibrium, hence inflating the discrepancy between active and passive cooling. Also, the amount of internal heat conversion affects the relative temperature lag. It appears that the larger the internal heat conversion, the smaller \( \Delta \tau \) becomes. Indeed, given the differential representation of the cooling law in Equation (15b) we see that the internal heat conversion can outweigh the radiative and convective
Figure 4: Relative time lag $\Delta \tau$ (Equation 18) for the internal heat conversion set to the maximum (a,c), and set to the minimum (b,d). The curves are generated for different equilibrium temperatures (see curve labels in °C). On the top row, the heating process is depicted (a,b), with the cooling process on the bottom row (c,d). The three vertical dashed lines represent typical surfaces for a SoC ($\approx 2.5$ cm$^2$), a smartphone ($\approx 70$ cm$^2$) and a tablet ($\approx 5$ dm$^2$). Data points on the right of the blue dashed lines have negative convective heat transfer coefficients.
coolings the larger it becomes. Thus the larger the internal heat conversion, the less sensitive the body becomes to changes in the radiative or convective cooling, and the more active and passive cooling will resemble.

5 Conclusion

We have developed the exact cooling law for passively cooled objects subject to radiation, (natural) convection, and internal heat conversion. The cooling law is shown to be analytically more complex than the commonly accepted active or exponential cooling law, which is technically sound for forcibly cooled objects. Unfortunately, the exact solution for the passively cooled object is a function of time: \( T(t) \). Attempting to invert the solution seems futile, so numerical approaches will have to be used when using radiation-aware thermal management units. A prototype implementation of the exact solution for passively cooled objects is presented.

Via computational fluid dynamics (CFD) and analytical simulations, we showed that the difference between active and passive cooling depends on three factors: 1) the surface size of the object, 2) the internal heat conversion, and 3) the equilibrium temperature. For small surfaces, e.g., SoC, we showed that the difference between active and passive cooling will be difficult to detect. For medium sized objects, depending on the magnitude of the internal heat conversion and equilibrium temperature, the discrepancy between active and passive cooling could tentatively go unnoticed. For large objects, large discrepancies may be present. We also highlighted that the quantization noise of temperature sensors may conceal temporal information between active and passive cooling. As the cooling law for passively cooled devices is quite elaborate to work with and the benefit of a scientifically sound cooling law is limited by lack of accurate temperature sensors, we can state that, for systems minimizing overhead, assuming an exponential cooling law will likely not induce large perceptual deviations from reality.

In this work we considered the cooling of an isothermal object. In practical situations this assumption doesn’t always hold. To obtain a more realistic model we need to consider internal conduction and hence also thermal hotspots. There impact on our heat model of these considerations is part of our future work. Moreover, in embedded systems, a microprocessor is usually mounted on a PCB and covered by other objects, such as an LCD display or others. The presence of these objects also interacts with the passive cooling of the microprocessor. Most likely numerical methods will have to be deployed to gain more understanding for such conditions.

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Appendices

A Applicability of the Passive Heat Equation

Previously we assumed that the internal heat conversion \( H(T) \) was a linear function, i.e., polynomial of the first-order with coefficients elements of \( \mathbb{R}^+ \). Given that the radiation absorbed or emitted by a body is described by a 4th-order polynomial, we discuss the implications of an arbitrary \( H(T) \) up to the 3th-order. We will show via logical reasoning that the analytic solution in the paper holds for \( H(T) \) up to the 4th order under certain conditions.

Let us define a body that is radiating energy at a rate \(-\delta\), and subject to other heat transfer mechanisms described by a polynomial \( K(T) \), e.g., internal heat conversion. Let \( K(T) \) be a polynomial of an order not larger than three. Then the thermal energy storage rate into the body is equal to:

\[
C \frac{dT}{dt} = -\delta T^4 + K(T) = -\delta T^4 + (\kappa_3 T^3 + \kappa_2 T^2 + \kappa_1 T + \kappa_0),
\]

(19)
Figure 5: Visualization of Equation (20) for several variations of the right-hand side polynomial ($K(T)$). Polynomials: 1st order (green, black), 2nd order (red, gray), 3rd order (purple), and $\delta T^4$ (blue). The black bullets represent the intersections of each polynomial with $\delta T^4$.

where we define $\delta \in \mathbb{R}_0^+, \kappa_{0,1,2,3} \in \mathbb{R}$, and $C$ is the thermal capacity of the system. $\delta$ must be positive as $-\delta T^4$ represents the heat emitted by the body via radiation. $\kappa_{0,1,2,3}$ are the constants of a polynomial describing the function $K(T)$. To solve the differential in Equation (19) the roots need to be found. In particular, we have solved the differential equation for a 4th-order polynomial assuming two real and two complex conjugate roots. To find the roots of Equation (19) we evaluate it at the equilibrium temperature $T = T_e$, then $dT/dt = 0$:

$$\delta T^4 = \kappa_3 T^3 + \kappa_2 T^2 + \kappa_1 T + \kappa_0.$$  (20)

This equality is visualized in Figure 5. There the solid blue curve represents the contribution on the left-hand side and the other dashed lines are possible examples of the polynomial in the right-hand side. It can be seen that it is easy to construct polynomials that have one or two intersections with $\delta T^4$. Also curves can be constructed that intersect the $\delta T^4$ only in one point (for example the dashed black line in Figure 5); such points are counted as two roots. The dashed gray line is an example of a polynomial without any intersection with $\delta T^4$. Only those polynomials with one or two intersections with $\delta T^4$ have physical meaning in the context discussed in this paper. One or two intersections with $\delta T^4$ produce two real roots and two complex conjugate roots. No intersections with $\delta T^4$ would imply that there exists no equilibrium temperature, i.e., the system is not thermally stable.

**B Solving the Passive Heat Equation**

The differential formulation of a passively cooled object with linear internal heat conversion can be described as follows, as per Equation (9):

$$\frac{dT}{dt} = \frac{1}{C}(-\varepsilon \sigma S T^4 + (\eta_1 - hS)T + (\eta_0 + S(hT_a + \varepsilon \sigma T_a^4))).$$  (21)

The right-hand side is a fourth-order polynomial. The previous equality can be rephrased as:

$$\frac{dT}{dt} = -\kappa_4 T^4 + \kappa_3 T^3 + \kappa_2 T^2 + \kappa_1 T + \kappa_0,$$  (22)

where the constants $\kappa_4 \in \mathbb{R}_0^+$ and $\kappa_{0,1,2,3} \in \mathbb{R}^+$. Rearranging this equation yields

$$\int \frac{1}{T^4 - \frac{\kappa_4}{\kappa_1} T^3 - \frac{\kappa_3}{\kappa_1} T^2 - \frac{\kappa_2}{\kappa_1} T - \frac{\kappa_1}{\kappa_1}} dT = -\kappa_4 \int dt.$$  (23)
The integration of the fraction on the left-hand side can be achieved via partial fractions decomposition:

\[
\int \frac{1}{T^4 - \frac{\kappa_4}{\kappa_4} T^3 - \frac{\kappa_4}{\kappa_4} T^2 - \frac{\kappa_4}{\kappa_4} T - \frac{\kappa_4}{\kappa_4}} dT = \int \frac{1}{(T - \omega_1)(T - \omega_2)(T - \omega_3)(T - \omega_4)} dT \tag{24}
\]

The roots \( \omega_i \) of the 4th order polynomial in the denominator can be obtained via Ferrari’s theorem \[23\], and other methods such as Newton's and the secant. Given that there exist a maximum of one or two real unique values for \( T \) that satisfy

\[
\kappa_4 T^4 = \sum_{i=0}^{3} \kappa_i T^i,
\]

we can state that two roots are real, say \( \omega_{(1,2)} \); the other two roots are complex conjugates\(^1\). This means that \( \Re(\omega_3) = \Re(\omega_4) \) and \( \Im(\omega_3) = -\Im(\omega_4) \), which simplifies a few things. As the initial differential equation is real, we are looking for a real solution too; thus the imaginary part must equate to zero. This is however automatically taken care of as the product of the two complex roots yield a real sum:

\[
\frac{1}{(T - \omega_1)(T - \omega_2)} = \frac{1}{(T - (\Re(\omega_3) + i\Im(\omega_3)))(T - (\Re(\omega_4) - i\Im(\omega_4)))} = \frac{1}{(T - \Re(\omega_3))^2 + \Im(\omega_3)^2}.
\]

Whence, Equation \[24\] becomes

\[
\int \frac{1}{T^4 - \frac{\kappa_4}{\kappa_4} T^3 - \frac{\kappa_4}{\kappa_4} T^2 - \frac{\kappa_4}{\kappa_4} T - \frac{\kappa_4}{\kappa_4}} dT = \int \frac{A}{(T - \omega_1)} + \frac{B}{(T - \omega_2)} + \frac{CT + D}{(T - \Re(\omega_3))^2 + \Im(\omega_3)^2} dT. \tag{25}
\]

Henceforth we define \( \alpha = \Re(\omega_3) \) and \( \beta = \Im(\omega_3) \), and the following equality must be solved to obtain the values of \( A, B, C \) and \( D \):

\[
\frac{1}{(T - \omega_1)(T - \omega_2)((T - \alpha)^2 + \beta^2)} = \frac{A}{(T - \omega_1)} + \frac{B}{(T - \omega_2)} + \frac{CT + D}{(T - \alpha)^2 + \beta^2}, \tag{26}
\]

which can be expressed as a system of equations:

\[
\begin{align*}
0 &= A + B + C \\
0 &= D - \omega_1(B + C) - \omega_2(A + C) - 2\alpha(A + B) \\
0 &= \alpha^2(A + B) + \beta^2(A + B) + 2\alpha(\omega_2 A + \omega_1 B) - (\omega_1 + \omega_2)D + \omega_1\omega_2 D \\
1 &= -\alpha^2(\omega_2 A + \omega_1 B) - \beta^2(\omega_2 A + \omega_1 B) + \omega_1\omega_2 D
\end{align*}
\]

and can be solved via Gaussian elimination:

\[
\begin{bmatrix}
0 \\
0 \\
1
\end{bmatrix} = \begin{bmatrix}
1 & -\omega_2 - 2\alpha & -\omega_1 - 2\alpha & -\omega_1 - \omega_2 & 1 \\
0 & \alpha^2 + \beta^2 + 2\alpha\omega_2 & \beta^2 + 2\alpha\omega_1 + \alpha^2 & -\omega_1 - \omega_2 & \omega_1\omega_2 \\
0 & -\alpha^2\omega_2 - \beta^2\omega_2 & -\alpha^2\omega_2 - \beta^2\omega_1 & -\alpha^2\omega_1 - \beta^2\omega_1 & 0
\end{bmatrix} \times \begin{bmatrix}
A \\
B \\
C \\
D
\end{bmatrix}. \tag{27}
\]

So we obtain the expressions for \( A, B, C \) and \( D \):

\[
\begin{align*}
A &= \frac{1}{(\omega_1 - \omega_2)((\alpha^2 + \beta^2) - \omega_1(2\alpha - \omega_1))} \tag{28a} \\
B &= -A \frac{\alpha^2 + \beta^2 - \omega_1(2\alpha - \omega_1)}{\alpha^2 + \beta^2 - \omega_2(2\alpha - \omega_2)} \tag{28b} \\
C &= -(A + B) \tag{28c} \\
D &= A(2\alpha - \omega_1) + B(2\alpha - \omega_2) \tag{28d}
\end{align*}
\]

\(^1\)Appendix \[A\] shows that for our applications this is the case.
Continuing with Equation 25, this yields:

\[
\int \frac{CT + D}{(T - \alpha)^2 + \beta^2} dT = \int \frac{A(T - \alpha)}{(T - \omega_1)(T - \omega_2)} + \frac{CT + D}{(T - \alpha)^2 + \beta^2} dT
\]

\[
= \int \frac{A}{(T - \omega_1)} dT + \int \frac{B}{(T - \omega_2)} dT + \int \frac{CT + D}{(T - \alpha)^2 + \beta^2} dT
\]

\[
= A \ln |T - \omega_1| + B \ln |T - \omega_2| + \int \frac{CT + D}{(T - \alpha)^2 + \beta^2} dT + c_o,
\]

where \(c_o\) is an integration constant. The last term on the right-hand side may be integrated via substitution, where \(u = (T - \alpha)^2\), yielding \(du = 2(T - \alpha)dT\), and also \(v = \frac{T - \omega_a}{\beta}\), giving \(dv = \frac{1}{\beta}dT\):

\[
\int \frac{CT + D}{(T - \alpha)^2 + \beta^2} dT = \int \frac{C(T - \alpha)}{(T - \alpha)^2 + \beta^2} dT + \int \frac{\alpha C + D}{(T - \alpha)^2 + \beta^2} dT
\]

\[
= \frac{C}{2} \int \frac{2(T - \alpha)}{(T - \alpha)^2 + \beta^2} dT + \int \frac{\alpha C + D}{\beta^2} \int \frac{1}{(v^2 + 1)} dv
\]

\[
= \frac{C}{2} \ln |u + \beta^2| + \frac{\alpha C + D}{\beta} \arctan(v) + c_1
\]

\[
= \frac{C}{2} \ln |(T - \alpha)^2 + \beta^2| + \frac{\alpha C + D}{\beta} \arctan \left( \frac{T - \alpha}{\beta} \right) + c_1.
\]

where \(c_1\) is an integration constant. Then the solution to Equation 25 is as follows

\[
\int \frac{1}{T^4 - \frac{\alpha a}{\kappa_4} T^3 - \frac{\alpha c}{\kappa_4} T^2 - \frac{\alpha i}{\kappa_4} T - \frac{a o}{\kappa_4}} dT = A \ln |T - \omega_1| + B \ln |T - \omega_2| + \frac{C}{2} \ln |(T - \alpha)^2 + \beta^2|
\]

\[
+ \frac{\alpha C + D}{\beta} \arctan \left( \frac{T - \alpha}{\beta} \right) + c_o,
\]

(29)

where \(A, B, C\) and \(D\) are given in Equations 28 and \(\omega_\ast\) are the real roots of the polynomial in the denominator on the left-hand side, \(\alpha = \Re(\omega_3), \beta = \Im(\omega_3)\), and \(c_o\) is a (new) integration constant satisfying the initial conditions \(T(0) = T_0\).

Now we can complete Equation 23

\[
- \frac{1}{\kappa_4} \int dt = \int \frac{1}{T^4 - \frac{\alpha a}{\kappa_4} T^3 - \frac{\alpha c}{\kappa_4} T^2 - \frac{\alpha i}{\kappa_4} T - \frac{a o}{\kappa_4}} dT
\]

\[
t = - \frac{1}{\kappa_4} \left( A \ln |T - \omega_1| + B \ln |T - \omega_2| + \frac{C}{2} \ln |(T - \alpha)^2 + \beta^2|
\]

\[
+ \frac{\alpha C + D}{\beta} \arctan \left( \frac{T - \alpha}{\beta} \right) + c_o \right),
\]

(30)

Surprisingly, our result is the same solution as presented by Besson [24], even though he modeled a different, but similar, physical problem. Besson however assumed some simplifications, different assumptions from ours, and solved the differential equation via other methods. Nonetheless his solution also contains three logarithms, one of them containing a second-order polynomial, and an arctan. Because of Besson’s simplifying assumptions, however, his equation is limited to the case where \(T - T_a = T + \frac{m a}{\kappa_4}\), which is a special case of our initial problem.

C Derivations of Approximations for \(f(T) = t\)

The exact passive cooling law as presented is of the form \(f(T) = t\). For practical reasons we desire a formulation of the form \(f(t) = T\). Unfortunately inverting the exact passive heat equation is challenging. We develop three approximations to the exact passive cooling law which are more easily invertible.
C.1 The Quadratic Approximation

Stefan-Boltzmann’s law of radiation states that the energy emitted by radiation is proportional to \( T^4 \) (Equation 2). Because of this term the polynomial of Equation 10 is of the fourth-order. More specifically, it is the two imaginary roots of the fourth order polynomial that introduce the \( \arctan \) in Equation 11. If we were to approximate \( \arctan \) straightforward. The quadratic approximation

\[
T^4 = q_0 + q_1T + q_2T^2
\]

introduces an error between -0.041% and 0.072% for 20°C < \( T < 65°C \), which is very acceptable. Then the quadratic approximation to Equation 22 would be equal to solving

\[
\frac{dT}{dt} = \kappa_2 T^2 + \kappa_1 T + \kappa_0.
\]

The solution to this equation, assuming two real roots (\( \omega = (-\kappa_1 \pm \sqrt{\kappa_1^2 - 4\kappa_2\kappa_0}/(2\kappa_2)) \)) and that \( \kappa_2 < 0 \):

\[
t = -\frac{1}{\kappa_2} \left( A \ln |T - \omega_1| + B \ln |T - \omega_2| + c_o \right),
\]

where \( A = 1/(\omega_2 - \omega_1) \) and \( B = -A \). Now we can isolate \( T \) as follows:

\[
t + \frac{c_o}{\kappa_2} = -\frac{A}{\kappa_2} \left( \ln |T - \omega_1| - \ln |T - \omega_2| \right)
\]

\[
-\frac{\kappa_2 t + c_o}{A} = \ln \left( \frac{|T - \omega_1|}{|T - \omega_2|} \right)
\]

\[
c_o e^{-\frac{\kappa_2}{A} t} = \frac{|T - \omega_1|}{|T - \omega_2|}.
\]

Let’s define \( \omega_1 \) and \( \omega_2 \) such that \( \omega_1 < \omega_2 \). As we are operating in the temperature range 0°C < \( T < 100°C \) and given the shape of the quadratic approximation, \( T \) will always be larger than \( \omega_1 \). Hence we can assume that \( T - \omega_1 > 0 \). The absolute value of \( T - \omega_2 \) forces us to distinguish two cases, i.e. where \( T > \omega_2 \) and the case for \( T < \omega_2 \). Bear in mind that \( \omega_2 \) is also the equilibrium temperature \( T_e \) of the system. This corresponds either to the heating or the cooling process, respectively. Let’s look at the cooling process first. We have \( T > \omega_2 \), then

\[
T - \omega_1 = (T - \omega_2)c_o e^{-\frac{\kappa_2}{A} t}
\]

\[
T - e_o e^{-\frac{\kappa_2}{A} t} = \omega_1 - \omega_2 e_o e^{-\frac{\kappa_2}{A} t}
\]

\[
T = \frac{\omega_1 - \omega_2 e_o e^{-\frac{\kappa_2}{A} t}}{1 - c_o e^{-\frac{\kappa_2}{A} t}}
\]

and accordingly for \( T < \omega_2 \), or the heating process, we get:

\[
T = \frac{\omega_1 + \omega_2 e_o e^{-\frac{\kappa_2}{A} t}}{1 + c_o e^{-\frac{\kappa_2}{A} t}},
\]

where \( c_o \) is an integration constant to meet the initial condition \( f(0) = T_0 \), and given by

\[
c_o = \frac{|T_0 - \omega_1|}{|T_0 - \omega_2|}.
\]

The roots \( \omega_1 \) are easily found as follows:

\[
\omega_1 = -\frac{\kappa_1 + \sqrt{\kappa_1^2 - 4\kappa_2\kappa_0}}{2\kappa_2} \quad \text{and} \quad \omega_2 = -\frac{\kappa_1 - \sqrt{\kappa_1^2 - 4\kappa_2\kappa_0}}{2\kappa_2}.
\]
The equilibrium temperature $T_e$ is defined by the positive root $\omega_2$. A prototype implementation of Equation \[31\] is provided in Appendix \[D.2\].

In the above derivation, we have fixed the coefficients $q_*$ in Equation \[31\]. These values were chosen to fit best in a certain temperature range. To be more universally applicable, however, the coefficients could be generated dynamically such that they are optimally tailored to the temperature range of concern.

### C.2 The First Order O’Sullivan Approximation

O’Sullivan [25] presented an approximation for a cooling law including convection and radiation, but without the presence of internal heat conversion. Though, we can extend his approximation with internal heat conversion. We will use an alternative formulation of the internal heat conversion such that we can more easily apply our variable substitution later on: $H(T) = \eta T + \eta_0 = \eta_1 (T - T_a) + \eta_1 T_a + \eta_0$. The initial definition of the passive heat Equation \[9\] is then given by:

\[-C \frac{dT}{dt} = \epsilon \sigma S(T^4 - T_a^4) + hS(T - T_a) - (\eta_1 (T - T_a) + \eta_1 T_a + \eta_0)\]

Let’s introduce the variable $\theta = T - T_a$:

\[-C \frac{d\theta}{dt} = \epsilon \sigma S((\theta + T_a)^4 - T_a^4) + (hS - \eta_1) \theta - (\eta_1 T_a + \eta_0)\]

Now, we can apply binomial expansion\(^2\) to $(\theta - T_a)^4$, whence:

\[-C \frac{d\theta}{dt} = \epsilon \sigma S(\theta^4 + 4T_a \theta^3 + 6T_a^2 \theta^2 + 4T_a^3 \theta + T_a^4) - T_a^4) + (hS - \eta_1) \theta - (\eta_1 T_a + \eta_0)\]

\[-C \frac{d\theta}{dt} = \epsilon \sigma S(\theta^4 + 4T_a \theta^3 + 6T_a^2 \theta^2) + (hS - \eta_1 + 4\epsilon \sigma ST_a^3) \theta - (\eta_1 T_a + \eta_0)\]

\[-C \frac{d\theta}{dt} = k \theta^4 + l \theta^3 + m \theta^2 + n \theta + p\]

where the coefficients for surfaces around $1 \text{ dm}^2$ are as follows:

\[
\begin{align*}
  k &= \epsilon \sigma S \sim 10^{-10} \\
  l &= 4\epsilon \sigma ST_a \sim 10^{-7} \\
  m &= 6\epsilon \sigma ST_a^2 \sim 10^{-5} \\
  n &= (hS - \eta_1 + 4\epsilon \sigma ST_a^3) \sim 0.01 \\
  p &= -(\eta_1 T_a + \eta_0) \sim 1
\end{align*}
\]

Now, if $(T - T_a)$ is not too large the series on the right-hand side of Equation \[35\] converges reasonably fast [25]. Depending on the accuracy desired, the higher-order coefficients may be dropped. Let’s see how well a first-order and a second-order approximation behaves. As expected, the first-order approximation yields also an exponential law:

\[-C \frac{d\theta}{dt} = n \theta + p \\
\int \frac{1}{\theta + p/n} d\theta = -\frac{n}{C} \int dt \\
\ln(\theta + p/n) = -\frac{n}{C} t + c_o \\
\theta = c_o e^{-p/n} - \frac{p}{n}\]

where $c_o$ is an integration constant such that $\theta(t = 0) = T_0 - T_a$:

\[c_o = \theta_0 + \frac{p}{n} = (T_0 - T_a) + \frac{p}{n}\]

\(^2\)Binomial expansion of the 4th order: $(x + y)^4 = x^4 + 4x^3y + 6x^2y^2 + 4xy^3 + y^4$. 
And so the first-order O’Sullivan solution is:

\[ T = \left( T_0 - T_a + \frac{p}{n} \right) e^{-\frac{n}{n}} - \frac{p}{n} + T_a. \] (36)

Astonishingly, this approximation is a pure exponential function. The equilibrium temperature \( T_e \) and, accordingly, the convective heat transfer coefficient \( h \) at \( T_e \) are given by:

\[
T_e = -\frac{\eta_1 T_a + \eta_0}{hS - \eta_1 + 4\epsilon \sigma T^3} + T_a
\]

\[
h = \frac{\eta_0 - \eta_1 T_a}{S(T_e - T_a)} + \frac{\eta_1}{S} - 4\epsilon \sigma T^3
\]

A prototype implementation of Equation 36 is provided in Appendix D.2.

C.3 The Second-Order O’Sullivan Approximation

The second-order O’Sullivan approximation is a bit more complex compared to the first-order O’Sullivan approximation. Moreover, the derivation looks also significantly different from the original derivation of O’Sullivan [25], given the presence of the constant term \( p \) in Equation 35.

The second-order O’Sullivan approximation is similar to the coefficient approximation in the sense that solving

\[-C \frac{d\theta}{dt} = m\theta^2 + n\theta + p\] (37)

is similar to solving Equation 32. Thus the solution for the second-order O’Sullivan approximation will be the same as for the quadratic approximation, except for the constants. We can thus state that the second-order O’Sullivan approximation is given by:

\[ T = \frac{\omega_1 \pm \omega_2 e^{-\frac{\omega_1}{n} t}}{1 \pm e^{-\frac{\omega_2}{n} t}} + T_a, \] (38)

where ”±” becomes ”+” for \( T_e > T_0 \), and ”−” for \( T_e < T_0 \). \( \omega_1 \) is given by:

\[
\omega_1 = -\sqrt{n^2 - 4pm - n^2} \text{ and } \omega_2 = \sqrt{n^2 - 4pm - n}. \]

The constant \( A \) and integration constant \( c_o \) such that \( \theta(0) = \theta_0 \):

\[
A = -\frac{1}{\omega_2 - \omega_1} \text{ and } c_o = \frac{\theta_0 - \omega_1}{|\theta_0 - \omega_2|},
\]

where \( \theta_0 = T_0 - T_a \). The equilibrium temperature \( T_e \) is defined by \( \omega_2 + T_a \).

A prototype implementation of Equation 38 is provided in Appendix D.2.

D Prototype Implementations of the Passive Cooling Law

D.1 Exact Passive Cooling Law

A prototype implementation of the passive cooling law as per Equation 11 is presented for R. The function \texttt{passive_cooling_coefs} computes the coefficients of the passive cooling law given a temperature \( T_0 \) at \( t=0 \). \( a, b, c, d \) and \( e \) are the coefficients of the polynomial on the right-hand side of Equation 10. The function \texttt{passive_cooling} computes the time at which the system attains the temperature \( T_e \), given the coefficients computed by \texttt{passive_cooling_coefs}.

\[
\texttt{passive_cooling_coefs <- function(a,b,c,d,e,T0) \{
my_roots <- polyroot(c(e/a,d/a,c/a,b/a,1))
my_complex <- apply(as.matrix(my_roots), 2, function(x) {return(abs(Im(x)) > 1e-3)})
my_real <- which(my_complex == FALSE)
my_imag <- which(my_complex == TRUE)
\}\}
\]

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omega <- Re(my_roots[my_real])[order(Re(my_roots[my_real]))]
alpha <- Re(my_roots[my_imag][1])
beta <- Im(my_roots[my_imag][1])

A <- 1/((omega[1]-omega[2])*(alpha^2+beta^2)-omega[1]*(alpha*2-omega[1]))
B <- -1/((omega[1]-omega[2])*(alpha^2+beta^2)-omega[2]*(alpha*2-omega[2]))
C <- -(A+B)
D <- A*(alpha*2-omega[1])+B*(alpha*2-omega[2])

co <- -(A*log(abs(T0-omega[1]))
 +B*log(abs(T0-omega[2]))
 +C/2*log(abs((T0-alpha)^2+beta^2))
 +(alpha*C+D)/beta*atan((T0-alpha)/beta)
 +co)

output <- rbind(c(a,b,c,d,e,A,B,C,D,omega[1],omega[2],alpha,beta,co,abs(omega[2])))
colnames(output) <- c("a","b","c","d","e","A","B","C","D",
 "w1","w2","alpha","beta","co","Te")
return(output)
}

passive_cooling <- function(coefs, T) {
a <- coefs[,"a"]
A <- coefs[,"A"]
B <- coefs[,"B"]
C <- coefs[,"C"]
D <- coefs[,"D"]
omega <- c(coefs[,"w1"],coefs[,"w2"])
alpha <- coefs[,"alpha"]
beta <- coefs[,"beta"]
co <- coefs[,"co"]

return(1/a*(A*log(abs(T-omega[1]))
 +B*log(abs(T-omega[2]))
 +C/2*log(abs((T-alpha)^2+beta^2))
 +(alpha*C+D)/beta*atan((T-alpha)/beta)
 +co))
}

A realistic example of the use of our passive cooling algorithm goes as follows:

my_coefs <- passive_cooling_coefs(-3.56e-11,0,0,9e-4, 7.07e-1,400)
my_temperature <- seq(400,438,0.1)
my_time <- passive_cooling(my_coefs, my_temperature)
plot(x=my_time, y=my_temperature, type="l", col="red")
abline(h=my_coefs[,"Te"], v=0, col="black", lty="dashed")

The resulting graph is shown in Figure 6.

D.2 Approximations to the Exact Passive Cooling Law

The following functions approximate the exact cooling law. They are formulated as \( f(t) = T \). The following arguments are used: \( e \) is the emissivity of the object, \( h \) is the convective heat transfer coefficient, \( S \) the surface area, \( Ta \) the ambient temperature, \( C \) the thermal capacity of the object, \( eta1 \) and \( eta2 \) coefficients describing the internal heat conversion, and \( T0 \) is the temperature of the object at \( t=0 \).

D.2.1 Coefficient Approximation

parisTechStefanAnalyticalApproxRadiation <- function(t, e, h, S, Ta, C, eta1, eta2, T0) {
k4 <- -emission*sigma*S/C
Figure 6: Resulting plot that is generating when executing the prototype implementation in R.

```r
k1 <- (eta1 - h*S)/C
k0 <- ((h*S*Ta + e*sigma*S*Ta^4 - eta1) + eta0)/C
x <- k4*598262
y <- -k4*2514836262 + k1
z <- k4*29700057265 + k0
omega1 <- (-y + sqrt(y^2 - 4*x*z))/(2*x)
omega2 <- (-y - sqrt(y^2 - 4*x*z))/(2*x)
A <- 1/(omega2 - omega1)
B <- -A
c0 <- abs(T0 - omega1)/abs(T0 - omega2)
if(omega2 > T0) {
  return((omega1 + omega2*c0*exp(-(t*x)/A))/((1 + c0*exp(-(t*x)/A))))
} else {
  return((omega1 - omega2*c0*exp(-(t*x)/A))/((1 - c0*exp(-(t*x)/A))))
}
```

D.2.2 First-order O’Sullivan Approximation

```r
approximation_osullivan_first_order <- function(t, e, h, S, Ta, C, eta0, eta1, T0) {
  sigma <- 5.6704e-8
  kelvin <- 273.15
  k <- e*sigma*S
  l <- 4*e*sigma*S*Ta
  m <- 6*e*sigma*S*Ta^2
  n <- h*S - eta1 + 4*e*sigma*S*Ta^3
  p <- -((eta1*Ta + eta0)
  return((T0 + (-Ta + p/n))*exp(-n/C*t) + (-p/n*Ta))
}
```
D.2.3 Second-order O’Sullivan Approximation

approximation_osullivan_second_order <- function(t, e, h, S, Ta, C, eta1, eta0, T0) {
  sigma <- 5.6704e-8
  kelvin <- 273.15
  k <- e*sigma*S
  l <- 4*e*sigma*S*Ta
  m <- 6*e*sigma*S*Ta^2
  n <- h*S - eta1 + 4*e*sigma*S*Ta^3
  p <- -(eta1*Ta + eta0)

  omega1 <- (-n-sqrt(n^2-4*m*p))/(2*m)
  omega2 <- (-n+sqrt(n^2-4*m*p))/(2*m)
  theta0 <- T0-Ta
  A <- -1/(omega2-omega1)
  B <- -A
  co <- abs(theta0-omega1)/abs(theta0-omega2)

  if((omega2+Ta) > T0) {
    return( (omega1+omega2*co*exp(-m/(C*A)*t))/((1+co*exp(-m/(C*A)*t)))+Ta )
  } else {
    return( (omega1-omega2*co*exp(-m/(C*A)*t))/((1-co*exp(-m/(C*A)*t)))+Ta )
  }
}

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