TERRESTRIAL CONSEQUENCES OF SPECTRAL AND TEMPORAL VARIABILITY IN IONIZING PHOTON EVENTS

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ABSTRACT

Gamma-ray bursts (GRBs) directed at Earth from within a few kiloparsecs may have damaged the biosphere, primarily through changes in atmospheric chemistry that admit greatly increased solar UV. However, GRBs are highly variable in spectrum and duration. Recent observations indicate that short (\(~0.1\) s) burst GRBs, which have harder spectra, may be sufficiently abundant at low redshift that they may offer an additional significant effect. A much longer timescale is associated with shock breakout luminosity observed in the soft X-ray (\(~10^3\) s) and UV (\(~10^5\) s) emission and radioactive decay gamma-ray line radiation emitted during the light-curve phase of supernovae (\(~10^7\) s). Here, we generalize our atmospheric computations to include a broad range of peak photon energies and investigate the effect of burst duration while holding total fluence and other parameters constant. The results can be used to estimate the probable impact of various kinds of ionizing events (such as short GRBs, X-ray flashes, and supernovae) on the Earth’s atmosphere. We find that the ultimate intensity of atmospheric effects varies only slightly with burst duration from \(10^{-1}\) to \(10^8\) s. Therefore, the effect of many astrophysical events causing atmospheric ionization can be approximated without including time development. Detailed modeling requires specification of the season and latitude of the event. Harder photon spectra produce greater atmospheric effects for spectra with peaks up to about 20 MeV because of greater penetration into the stratosphere.

Subject headings: astrobiology — gamma rays: bursts

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1. INTRODUCTION

Thorsett (1995) first pointed out the possible devastating impact of a gamma-ray burst (GRB) from within our Galaxy on the Earth. Scalo & Wheeler (2002) refined this expectation and made quantitative estimates of effects. Smith et al. (2004) did an explicit radiative transfer calculation of effects on the biosphere, as well as on Mars-like planets. More recently, we have proposed a possible connection with a particular mass extinction event (Melott et al. 2004) and computed in some detail the long-term atmospheric effects and some biogeophysical consequences of a “typical” long-duration GRB, which has a softer spectrum (long-soft burst [LSB]) beamed at the Earth from 2 kpc, a conservatively estimated probable nearest event in the last billion years (Thomas et al. 2005a, 2005b; Melott et al. 2005).

The greatest effect of the GRB photons is to make changes in atmospheric chemistry that generate odd nitrogen compounds (NO, NO2, NO3, N2O5, HNO3, HO2NO2, ClONO2, and BrONO2). Those usually called NO (NO, NO2) catalyze the depletion of ozone (O3). O3 normally prevents about 90% of the solar UVB (280 – 315 nm) radiation from reaching the surface. UVB is extremely damaging to most organisms, particularly since it easily damages DNA as well as proteins. In Thomas et al. (2005b) effects were explored as a function of altitude, latitude, and time for a burst at a variety of fluences, burst seasons, and latitudes. Season and latitude of the burst event were found to considerably modulate the intensity of the atmospheric effects.

In Thomas et al. (2005b) the Band et al. (1993) GRB spectrum was used with a peak at 187.5 keV and duration of 10 s, typical LSB values. However, GRBs show both spectral and temporal variability. Burst duration ranges over several orders of magnitude, with an apparently entirely different class of short bursts, often of order 0.1 s. The peak of the Band et al. spectrum shows considerable variation, and short burst GRBs often have a “harder” spectrum, with considerable flux above several MeV (Cline et al. 2005); hereafter, we refer to this class as short-hard bursts (SHBs). The SHB class may well comprise two or more populations, one cosmological and one in “nearby” galaxies. Although long-duration LSB rates have clearly been declining with cosmic time, there many other kinds of objects, such as SHBs, supernovae, and soft X-ray flashes, which might have terrestrial impacts. There is good evidence for at least one very young GRB remnant in our Galaxy (Atoyin et al. 2006). For further discussion see Dermer & Atoyin (2006).

It is also important to consider the spectral variation, luminosity evolution, and temporal variability of LSBs, including possible effects of afterglow on an already-damaged terrestrial atmosphere. It is impossible to fully explore the large parameter space relevant to effects on the Earth. In this paper we explore the consequences of burst duration and photon energy, while holding other variables constant, in order to arrive at a more general understanding. As better rate data become available, this will help to clarify the relative effects of SHBs, as well as soft gamma repeaters (e.g., Gotz et al. 2006), postburst flares from GRBs (e.g., Falcone et al. 2006), X-ray flashes from supernovae (e.g., Campana et al. 2006; see also Scalo et al. 2003), possible large solar flares (Schaefer et al. 2000), long-duration sources like Cygnus X-3, radioactive decay gamma-ray line radiation emitted for \(~10^7\) s during the light curve phase of supernovae (e.g., Teegarden et al. 1989; Karam 2002), and other possible sources of high-energy photons on the Earth and Earth-like planets. As our results are computed based on the fluence at the Earth, they can be interpreted in terms of a variety of source energies and distances and used to estimate the likely effects of various classes of events. Of course, many of these are very low probability events.
but we include a study of a wide variety of photon energy and duration to understand the dependence of the atmospheric response on these parameters.

Gehrels et al. (2003) studied the effect of relatively nearby supernovae on the Earth, using a different spectral model. While for extremely nearby (~8 pc) supernovae the effects are strong, it is difficult to disentangle the trends due to various assumed spectra, durations, etc. For this reason we include in this study events of very long duration (up to $10^8$ s), while keeping the fluence and spectrum constant. In this way we can isolate the effects of burst duration and spectral variation from others in their impact on the atmosphere, which will help in understanding the effects of nearby supernovae. In future work we will model more realistic supernovae.

2. METHODS

2.1. Atmospheric Model

The atmospheric code used here is a two-dimensional (latitude, pressure) time-dependent model with 18 equal latitude bins and 58 log-pressure bins, corresponding to approximately 2 km spacing. The code uses a lookup table for photolysis; mixing and small-scale winds are empirical. Here, we do not repeat a much more extensive description contained in Thomas et al. (2005b, including their appendices and tables). The code is valid for atmospheric composition similar to the present Earth, which should be true for about the last 500 Myr, except for an interlude of very high oxygen, which ended at the Permian extinction (Berner et al. 2003). There are also differences in transport due to changes in, e.g., continent placement. As present Earth is highly north–south asymmetric, the importance of this effect has been tested (Thomas et al. 2005b) by the north–south symmetry of the results obtained; it is known that systematic differences due to continent placement, wind currents, etc. are small, particularly when compared with effectively random effects of latitude and time of year of a given burst.

It is important to note that the chemistry is handled in bulk; once the overall ionization is computed, 1.25 NO$_\text{y}$ per ion pair produced; NO$_\text{y}$ is a “family” of compounds whose ratios are empirically based. No ions are included as intermediaries; their effects are subsumed in the bulk production rate. Some additional effects might come from the effects of intermediaries such as N$_\text{y}$. It has been argued (Jackman et al. 1979) that this ion might boost NO$_\text{y}$ production substantially above about 80 km, which is essentially irrelevant to the question of O$_3$ depletion, and that below this the effects are probably negligible.

The code has been used to compute effects on the atmosphere of ionizing radiation due to recent solar flares (Jackman et al. 1996 and references therein) and possible nearby supernovae (Gehrels et al. 2003). As described in Thomas et al. (2005b), we use a version with a shortened time step in order to account for the rapid variability of GRBs; here, we have had to shorten the time step even more to deal with SHB. The shortened time step model was tested for consistency with earlier computations of the same burst duration. As before, after the burst, the values are input into a longer term model with a 1 day time step, which includes diffusion and transport.

The code does not self-consistently calculate heating effects, but we have estimated these not to exceed 10 K for both direct heating by the burst and subsequent changes in stratospheric temperature due to lack of UV absorption by O$_3$, so these effects should be small. More discussion of heating effects due to O$_3$ changes is presented in Thomas et al. (2005b).

2.2. Burst Models and Input Parameters

We have controls in order to disentangle and clarify the effects of varying spectral and temporal properties of the burst. We use a burst at N45° latitude at noon on the March equinox, which is near the median of resulting intensity of effects in the (season, latitude) parameter space. The total fluence is 100 kJ m$^{-2}$, which corresponded to the expected nearest LSB beamed at the Earth within the past billion years, based on rate estimates (Melott et al. 2004; Thomas et al. 2005b), although Dermer & Holmes (2005) estimate a somewhat closer event. There is of course considerable uncertainty in the fluence due to uncertainties in the low-redshift rate in galaxies like the Milky Way, as well as due to the statistics of random events. Note that varying the fluence in Thomas et al. (2005b) showed effects changing less than linearly; ozone depletion there could be crudely characterized as varying with the cube root of the fluence. Therefore, these uncertainties in fluence translate into less uncertainty in the likely effect on terrestrial planets. For example, using the thin disk galaxy approximation, events of fluence 10 kJ m$^{-2}$ would have 10 times the effective surface area of origin and be expected on average 10 times as often as events with 10 times more fluence. But their effect on the biosphere might be crudely characterized as about half as severe (see Thomas et al. 2005b; particularly their Figs. 17 and 18 for more detail). Thus, a much lower LSB event rate as advocated by some (Fruchter et al. 2006; Stanek et al. 2007) does not translate into proportionately lower rates of trauma to the biosphere.

We vary spectra and duration around this previously modeled, extensively studied choice of parameters. Our previous work used the Band et al. (1993) spectrum with a peak at 187.5 keV, a typical long-duration GRB value. One of our primary goals here is to vary the peak energy. We do this by continuing to use the same spectral fit but varying the peak energy by 1 order of magnitude over steps 1.875, 18.75, 187.5 keV and 1.875, 18.75, 187.5 MeV. As before, the parameters in the Band et al. fit are $\alpha = -0.8$, $\beta = -2.3$ (Preece et al. 2000). This spectral shape was used previously as a good fit to a typical LSB; it is now used solely to avoid varying more than one parameter as we investigate the effect of photon energy. While this spectral shape (see Fig. 1; note that we plot energy flux, not photons) is somewhat...
arbitrary when considering events other than LSB, any other shape would be equally arbitrary for generic events.

As we are considering considerably harder and softer spectra than before, a wider range of attenuation coefficients are needed than those present in the lookup table embedded in the code. We have used the National Institute of Standards and Technology (NIST) XCOM database, available online (Berger et al. 2005), based on a mixture of 79% N and 21% O, except for 100 eV to 1 keV, which is not present there. In this range we use the data from Plechaty et al. (1981). Using this, we are able to include effects of photons impacting the upper atmosphere with initial energies from 100 eV to 100 GeV. The atmospheric code includes more constituents, but their abundance is small enough so as not to affect the attenuation of high-energy photons.

Some limitations should be noted. The cross sections for elements in the XCOM database pertain to isolated neutral atoms and do not take into account molecular effects that modify the cross sections, especially in the vicinity of absorption edges. Relatively small cross sections, such as those for Delbruck scattering, two-photon Compton scattering or photomeson production, are not included. Also omitted is the nuclear absorption and re-emission, which in the giant-dipole resonance region from 5 to 30 MeV, can contribute a few percent to the total attenuation coefficient. However, pair production in the nuclear field, which can dominate above 30 MeV, is fully included (see, e.g., Groom 1998). Finally, XCOM does not calculate energy absorption coefficients that represent the conversion of photon energy to kinetic energy of secondary Compton, photo-, and pair electrons.

Although XCOM provides opacities/absorption coefficients for photons of a wide range of energies, propagation and Compton scattering at high energies, above 10 MeV or so, must be treated with care and should involve detailed radiative transfer calculations. The problem arises from the Klein-Nishina correction to Compton scattering at large energies. The scattering cross section is suppressed roughly as $\sim 1/E$ at large scattering angles. Thus, high-energy photons will be predominantly scattered in the forward or backward direction, which makes use of angle-averaged cross sections somewhat inaccurate in stratified atmospheres. Accurate radiative transfer computations have been done at lower energies by Smith et al. (2004). Implementation of similar computational techniques in the existing atmospheric code is not feasible at present. However, since the high-energy mass attenuation coefficients we use are checked by laboratory experiment, and this approximate treatment should not be off by more than a few percent (Hubbell et al. 1980), the results should be sufficiently accurate, and average over these effects. Since the cross section is fairly flat above a few MeV, we do not expect large changes in increasing energy beyond this level. However, this expectation needs to be checked.

GRBs exhibit considerable diversity of duration for the prompt emission, although in our earlier work we used a fiducial 10 s burst. Many other kinds of ionizing events are also possible, with new candidates appearing at least yearly. For these reasons it is important to understand the effect of energy incident on other timescales. The clearest and most straightforward way to accomplish this is to consider the same fluence delivered over other times. We have therefore examined the effect of our original 187.5 keV peak spectrum, with the total fluence delivered over 0.1, 10, $10^2$, $10^3$, $10^4$, $10^5$, and $10^6$ s, which more than covers the observed range of extrasolar high-energy sources observed from SHB through supernovae. We had previously shortened the time step of the initial burst code from 225 s used for solar flares (Jackman et al. 2001) to 1 s; we now shorten it to 0.01 s for the 1 and 0.1 s burst models, with consistency checks implemented. It is important to note that while varying the spectrum and duration of the bursts individually, we have held the total fluence at the Earth constant at 100 kJ m$^{-2}$. Our goal is not to model every possible type of event, but to unearth general principles that will help us to understand effects and estimate the effect of various phenomena, some of which have yet to be observed. We have not modeled events with greater fluence. They may happen but exceed the likely range of validity of our atmospheric code due to temperature and transport feedback effects (Thomas et al. 2005b). In the present case heating of the stratosphere due to the prompt flux is limited to about 10 K, and later temperature changes due to changes in the heating of the stratosphere from changed ozone concentration are of similar magnitude. Therefore, these effects, which are not included in the code, are small. Fortunately, events with fluences above 100 kJ m$^{-2}$, which we cannot reliably model, are, given present knowledge, improbable at the Earth on gigayear timescales.

GRBs may produce a substantial flux of very high energy cosmic rays (e.g., Waxman 2004; Dermer & Holmes 2005), and a nearby supernova certainly would. This could produce a substantial additional effect beyond that of the photons, modeled here. The size of the expected cosmic-ray flux is uncertain, and its propagation in the Galactic magnetic field and the Earth’s magnetosphere is strongly energy-dependent, and could be impulsive or diffusive (depending on cosmic-ray energy and characteristics of the local Galactic magnetic field) to the target planet. For these reasons we do not include this, although it clearly could have an impact comparable to that of the photons for increased stress on the biosphere.

3. RESULTS

3.1. The Unperturbed Ozone Layer

There are at least three possible effects on biota (discussed in Melott et al. 2005): increased UV due to depletion of the protective ozone layer, possible climate change caused by reduced sunlight due to absorption by NO$_2$ produced in the atmosphere, and nitrate deposition as it is removed by rainout in the form of dilute nitric acid (in order of estimated declining importance). These are all closely coupled; we describe results here ultimately in terms of the DNA damage due to solar UVB transmitted through the O$_2$-depleted atmosphere. This appears to be the dominant effect for terrestrial planets, although direct irradiation from the event itself may dominate for Mars-like and other planets with thin atmospheres (Smith et al. 2004). Understanding this effect requires attention to the structure of the normal atmosphere. In Figure 2 we show the structure of the ozone layer output from our unperturbed atmospheric model. This reproduces the structure of the atmosphere well, except for effects of anthropogenic compounds such as chlorofluorocarbons, which we have removed from the code. Normally, O$_3$ is produced as a side effect of photolysis near the equator and at high altitude; it is transported to the poles, where it tends to accumulate. Some asymmetry of latitude is visible in the plot, due to the present configuration of continents on the Earth as well as the latitude and time of year. The first of these is a limitation of our model when used for paleontological purposes, but is known to be small compared with the latter two causes mentioned (Thomas et al. 2005b).

3.2. Effect of Incident Ionizing Radiation

For photons over the wide range of energies we shall consider, the attenuation coefficients (Berger et al. 2005) are such that only a tiny fraction of the incident radiation would reach the ground for a terrestrial-type atmosphere (Smith et al. 2004). Nearly all...
the energy goes into atmospheric chemistry. The primary chemical effect of the incident radiation is to break the strong chemical bonds of O\textsubscript{3} and N\textsubscript{2}, making possible the formation of molecules that are normally present in very low abundances in the atmosphere. NO and NO\textsubscript{2} are in this class; they also catalyze the destruction of ozone. We have found that it takes nearly a decade for the atmosphere to recover from a burst, as they are rained out and things return to the former nearly steady state. O\textsubscript{3} normally shields the Earth from most of the solar UVB (280–315 nm). During the time O\textsubscript{3} is depleted, organisms on the surface will be exposed to drastically increased levels of UVB, which is normally about 90% absorbed by ozone. UVB is strongly absorbed by DNA, breaking chemical bonds and leading to cancer and mutations. UVB levels found as a consequence of previous burst models are far in excess of those known to cause mortality in marine organisms, as summarized in Thomas et al. (2005b). UVA (315–400 nm) is also known to cause significant biological damage but can also be helpful in photosynthesis, which complicates things. However, UVA levels will not be changed noticeably by O\textsubscript{3} depletion, but may be somewhat lowered by NO\textsubscript{2} absorption. The relative contribution of UVA and UVB to biological damage is controversial and varies with the kind of effect considered.

In order to induce this effect, the NO\textsubscript{2} compounds must reach the ozone; one of the questions answered by our previous work was whether the photons would reach deeply enough into the atmosphere to do so. They do for a typical LSB; but since cross sections vary with energy, this is something that may strongly modify the effect of other kinds of energetic events.

### 3.3. Odd Nitrogen and Ozone Production

Energy from a burst could reach the ground and be a modest threat for intense nearby events, but this is dwarfed by the long-term effects of solar UVB for terrestrial type atmospheres. Nearly all of the energy goes into ionizing, dissociating, and heating the atmosphere from the stratosphere on up. The greatest departure from the usual state is the breaking of the extremely strong N\textsubscript{2} triple bond. This bond strength is the reason organisms need to “fix” nitrogen to make it available, and it is the reason that the atmosphere does not normally contain a large abundance of odd nitrogen compounds. However, the irradiation we are considering will break this bond as well as that of O\textsubscript{2}. The result is shown in Figure 3, which is 1 day after switchover to the 1 day time step code, or 5 days after the burst (however, NO\textsubscript{3} densities are almost the same 1 week later). Plots of NO\textsubscript{3} as a function of latitude and log pressure (approximate altitude) are shown for the six peak photon energies from 1.875 keV to 187.5 MeV. Again, the total fluence is the same in all cases. The difference in results is striking; effects are much greater for harder spectra. This can be well understood as an effect of the attenuation lengths. Softer bursts interact much higher in the atmosphere.

Note by comparing Figures 2 and 3 that there is substantial overlap between the vertical distributions of ozone and NO\textsubscript{3}, even for our lowest energy photons, which increases with photon energy. Thus, the increase in effect with increased photon energy is not many orders of magnitude, although it is substantial.

Reaction rates are different at different pressure, but most importantly the destruction of NO\textsubscript{3} compounds by photolysis goes on more rapidly at higher altitude, where solar UV is much more intense. (In our 1 week plots [not shown] the only visible difference from 1 day is some reduction of NO\textsubscript{3} by photolysis at high altitude). At lower altitude, the unperturbed atmosphere is largely self-shielded against UVC and UVB. Ironically, existing ozone helps to shield the odd nitrogen, which will eventually catalyze its destruction. When upper atmosphere ozone is depleted, there is some increased synthesis of ozone in the lower atmosphere.

These results also serve as a check on the Thomas et al. (2005b) results, which contained attenuation data up to 10 MeV and assumed a spectral cutoff there but the same total fluence. With the extended spectrum here but the same (187.5 keV) peak energy, effects are only a few percent greater.

To summarize, the primary effect of increasingly hard incident spectra is that the photons penetrate much more deeply into the stratosphere, where they are able to produce a great deal more odd nitrogen compounds, coincident with the location of the solar UVB-shielding ozone. As we discuss later, transport may redistribute the NO\textsubscript{3} compounds, but is far too slow to compete in a major way with any processes of their creation or effect on ozone for many months.

### 3.4. Ozone Depletion

In previous work (Melott et al. 2005) we described three effects of potential importance to a terrestrial-type biosphere from an ionizing radiation burst. They are increased UVB due to ozone depletion, increased opacity in the visible due to NO\textsubscript{2} production, which might reduce surface temperatures, and nitric acid deposition at the surface. For burst fluences probable over geologic timescales, the ozone depletion is clearly of major importance, while the others are questionable. For this reason, we describe the ozone depletion effect; the others scale closely with it.

Figure 4a shows fractional change in ozone density one week after the onset of the 1 day time step code, or 11 days postburst. Much of the variation with incident spectral hardness is understandable as a consequence of the odd nitrogen production. For harder spectra, ozone depletion extends deeper into the stratosphere. There is also some ozone production in the upper troposphere and lower stratosphere, shown as a positive percentage change. This production is not a large mass, as ozone is initially not of high abundance at low altitudes. It results from the small amount of incident burst ionizing radiation that penetrates that...
far, combined with an enhanced solar flux. Note that our print version has a bimodal gray scale; both the ozone depletion at high altitude and the ozone production at lower altitude and latitude are shown as darkened areas for this figure.

Since percent change is plotted here, the biggest absolute effect is the depletion of the stratospheric ozone. The mass of ozone lost from the upper stratosphere exceeds the mass of ozone produced in lower stratosphere. The log-pressure, or approximately altitude, axis is somewhat visually misleading as to the size of the effect. For this reason in Figure 4b we use a linear pressure (approximately log-altitude) axis. It is clear here that the vertical mass of ozone remaining decreases strongly with spectral hardness as the peak moves up to about 20 MeV.

The time evolution of the global average ozone column density is shown for all our burst spectra in Figure 5. The lines are not labeled, as the trend is simply increased depletion for increasing peak photon energy. Again, as a check on the Thomas et al. (2005b) results, we find close to 33% peak mean column density for the (187.5 keV) case, in which about 30% was found there. The difference can be understood as our inclusion of the photons in the power law beyond 10 MeV, which penetrate the stratosphere.

The increase in ozone depletion with energy as plotted here includes a partially compensating effect, the increased synthesis of ozone at lower altitudes made possible by greater depletion (shown in Fig. 4). We believe it is relevant to include this lower altitude ozone, as it does contribute to the shielding column density against UVB at the surface.

Changes over the first few months in the lowest energy photon cases are a consequence of the higher altitude of energy deposition. In these cases, photolysis of the NO$_y$ products at high altitude results in their decline and a partial ozone recovery at these altitudes. On the other hand, in the highest energy cases ozone depletion continues to worsen at for several weeks. In these cases there is less energy deposition at high altitudes, but some reaches the lower stratosphere. NO densities of order $10^8$ at these altitudes lead to an ozone destruction timescale of days. While the NO is not noticeably transported, ozone depletion can be seen (not shown in Fig. 4) to worsen in the lower stratosphere, most strongly for the higher energy cases. The rate of this slow ozone depletion is (at high latitudes and low altitudes) competitive with the new rate of ozone synthesis, whereas at low latitudes, with a higher Sun angle, synthesis dominates at lower altitudes. The photolysis loss of NO$_y$ and slight ozone recovery dominates overall below the 187.5 keV energy; completion of the reaction and slight worsening of ozone depletion dominates overall above 187.5 keV. We do not find sufficient ozone synthesis for toxic effects.

Values for the vertical eddy diffusion coefficient ($K_z$) in the model are from Fleming et al. (1999). There is uncertainty in this quantity that may affect our results. We are primarily concerned

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**Fig. 3.**—Latitude vs. log-pressure distribution of the density per centimeter cubed of odd nitrogen compounds 5 days after (10 s) bursts at N45° with peak photon energies over 6 orders of magnitude, as described in the text. Higher energy photons penetrate deeper into the stratosphere, where the compounds produced are more protected from early destruction by photolysis. [See the electronic edition of the Journal for a color version of this figure.]
with how this transport may move NO\textsubscript{y} produced at higher altitudes (i.e., in lower peak energy bursts) down to where the greatest concentrations of O\textsubscript{3} occur.

The value of $K_z$ varies from 0.2 m\textsuperscript{2} s\textsuperscript{-1} at the bottom of the stratosphere to about 0.02 m\textsuperscript{2} s\textsuperscript{-1} between approximately 20 and 30 km altitude and then climbs again toward the lower stratosphere value (0.2 m\textsuperscript{2} s\textsuperscript{-1}) going upward (to about 40 km). The exact value is dependent on seasonal variations as well as latitude. For this discussion we use the annual average at the equator.

The distance a constituent is moved by this diffusion is given by $(K_z t)^{1/2}$. Taking $K_z = 0.15$ m\textsuperscript{2} s\textsuperscript{-1} (the value at about 40 km) and $t = 1$ day, the distance is 113 m. If the actual $K_z$ value should be a factor of 10 larger, then this distance is about 360 m.

Therefore, this process is not a major factor in moving NO\textsubscript{y} over tens of kilometers on timescales shorter than weeks to months. In addition, even a factor of 10 uncertainty does not greatly affect the results, particularly on short timescales. The more critical quantity to our results, not affected by this uncertainty, is transport across the boundary into the troposphere, where rainout can begin.

For all cases, the ozone depletion then remains approximately constant for about 2 yr. Some recovery is noticeable after 5 yr, and is effectively complete in a decade. This is many generations for the highly UV-transparent single-celled organisms that lie at the base of the marine food chain. Environmental consequences will be severe and more so for events with increasingly hard spectra.

3.5. DNA Damage

Depletion of atmospheric ozone will cause substantially more UVB to penetrate the atmosphere and irradiate organisms near the surface. UVB interacts strongly with protein and DNA molecules, damaging them. This can cause gross damage, as well as cancers and an increase in the mutation rate. Deep water organisms should be affected only indirectly, as UVB photons are absorbed by a few meters of water; however, they may suffer to the extent that most are dependent on a food chain that begins on and near the surface, with phytoplankton. There are a variety of ways to measure biological damage (as discussed in Thomas et al. 2005b). “Action spectra” are defined for various biological effects. The strong effects of ozone depletion are for those that peak in the UVB regime. There is increasing appreciation of the damaging effects of UVA, which may be competitive with UVB damage, and even some evidence for DNA damage in hamster cells with blue light (Kielbasa et al. 1997). On the other hand, UVA is beneficial for some photosynthetic organisms (Mengelt & Prézelin 2005). Only UVB will be substantially enhanced as a result of the O\textsubscript{3} depletion. In fact, UVA and blue/violet may be slightly reduced due to the opacity of brown NO\textsubscript{2} in this band. Since UVB will increase greatly, and UVA/violet/blue will be constant or slightly decreased, it is appropriate to focus on the
UVB damage. We follow Thomas et al. (2005b) in using the DNA damage action spectrum of Setlow (1974). This action spectrum is an average over many kinds of damage to E. coli, DNA, phages, etc., which do not lie more than a factor of 2 (usually much less) off Setlow’s fitted curve. We plot relative, not absolute effects, which reduces the sensitivity of our presentation to the action spectrum. Our proxy for DNA biological damage is the UVB radiation transmitted by the depleted atmosphere per day, convolved with this biological action spectrum, taking into account, for example, the Sun angle and length of the day. For this purpose, we do a full computation of the changed optical depth due to changed O3 in the postburst atmosphere in 4 nm wavelength bands over the range 280–315 nm. The solar radiation that is transmitted through this depleted atmosphere is convolved with the Setlow action spectrum and integrated over time, giving an illustrative UVB-based biological effect as a function of time and latitude. Of course, overall mutation and cancer rates may vary nonlinearly with the assumed DNA damage rates because of repair mechanisms that are not taken into account here. However, these plots should at least give a good approximate indication of the approximate increased damage to photosynthetic single-celled organisms which lie at the base of the marine food chain.

In Figure 6 we show the result of this computation as a function of time for a typical preburst year followed by the first 3 postburst years. There is considerable evidence (summarized in Melott et al. 2004; Thomas et al. 2005b) that increases of a few times 10% in this function can lead to severe damage and mortality for marine microorganisms. As expected, the results are more severe with increasing spectral hardness, up to a 20 MeV peak. We are not now modeling any particular observed event, but we may compare these results with those noted in Thomas et al. (2005b). An order-of-magnitude increase in fluence approximately doubled the ozone depletion in that study. We find here that an order-of-magnitude increase in photon energy substantially increases the amount of ozone depletion. For example, a 2 keV peak burst with fluence 100 kJ m\(^{-2}\) could be expected to do about as much damage as a 200 keV peak burst with fluence 10 kJ m\(^{-2}\).

There is a substantial differential in effect for harder spectra, which may partially compensate for lower luminosity or greater distance. Using the 1 yr postburst globally averaged ozone depletion values in Figure 5 to estimate the amount of UVB that would reach the surface compared with unperturbed values for the 100 kJ m\(^{-2}\) irradiation, one arrives at “average” values from 1.5 to 2.7 times normal. As can be seen, localized preburst values of the UVB flux under the biological weighting window never go much above twice the global mean, but postburst they reach local values from approximately 4 to 15 times this mean, depending on the spectrum. Since values of a few times 10% UVB increase are typically lethal for phytoplankton and similar organisms that lie at the base of the food chain, none of these are remotely “safe” for...
the biosphere. Although detailed modeling of individual events is needed, one can estimate an approximate level of effect by scaling from these results.

3.6. Trends with Burst Duration

Varying around our “standard” 100 kJ m\(^{-2}\) burst peaking at 187.5 keV, we have examined the ozone depletion history of seven simulations in which the energy is deposited over timescales ranging from 0.1 to 10\(^8\) s. The latter is longer than any plausible nearby event type (but might be relevant if an object such as Cygnus X-3 were within a few parsecs of a terrestrial planet). It is included to better elaborate trends. The following simple result may be the most useful in this paper; there is no major duration effect. The maximum global average ozone depletions range between 33% and 37%, there are no large differences in the duration of the effect or the time-integrated ozone deficit. This result makes possible evaluation of the atmospheric impact of ionizing photon events without detailed modeling of the history (e.g., GRB variability, supernova light curve) by simply using the integrated flux. To first order, the ozone depletion is limited by the manufacture of ionized and dissociated N and O, which is limited by the incident fluence.

Let us first discuss the physical basis of this result. It says that the fluence, not the flux, is the best estimator of the size of the effect in the duration range we have considered. It implies that to first order the effect of one photon is largely independent of the effect of another photon. Why is this true, and why do rate-limiting reactions have so little effect? We mention a number of general conditions and then move to some specifics.

First, the dissociation of N\(_2\) by photons and subsequent formation of reactive compounds must be more rapid than the removal of these compounds. This is easily met; the timescale of the former is seconds to minutes and that of the latter (by transport to the troposphere and rainout, mostly as nitric acid) is many years. Of course, irradiation for periods longer than this would obviously compete with rainout. Secondly, reaction of N to form NO\(_x\) must be faster than competing reactions, including photolysis. We find that photolysis is only important in the ionosphere and far upper stratosphere, above the O\(_3\) layer, and that its timescale is days to weeks. Other reactions (see below) also cannot compete with this primary channel.

The primary driver to modify atmospheric chemistry is atomic N. This can recombine through the two-body reaction

\[
N + N \rightarrow N_2 + \gamma. \tag{1}
\]

This is so severely constrained by simultaneous momentum and energy conservation that the available phase space is very small. We have not found any laboratory measurements of this rate; it is so slow that even at very low pressures it is superseded by interactions with container walls or with trace contaminants. It may become important in the upper ionosphere, at altitudes too high to affect our ozone results. A more relevant competitor is the three-body interaction

\[
N + N + M \rightarrow N_2 + M, \tag{2}
\]

where M is an arbitrary third body and may carry off kinetic energy. For the Earth’s atmosphere, N\(_2\) is the most plausible identity for M. Reactions for N\(^+\) have similar timescales. Making the most pessimistic assumptions (all interacting photons are instantaneously converted to N atoms), we find that this reaction is still about 25 orders of magnitude slower than the most important short-term, in situ reaction for the recombination of N,

\[
N + NO \rightarrow N_2 + O, \tag{3}
\]

which is included in our code, does have some effect, and whose effects compete slightly with the creation of O\(_3\)-depleting compounds as discussed below. Once newly formed N is removed by this or other (less significant) reactions, it is out of the depletion cycle.

Most of the products of ionization and dissociation begin their ozone-depleting interaction with the atmosphere by forming NO. The most important reaction is

\[
N + O_2 \rightarrow NO + O. \tag{4}
\]

NO\(_2\) is easily formed from NO. We do not repeat here the O\(_3\) depletion catalysis cycle, except to mention that it involves primarily NO and NO\(_2\), which are recycled and not depleted by it.

We can discuss the relative importance of reactions (3) and (4) given abundances in the normal or postburst atmosphere and rate constants given in Thomas et al. (2005b). The timescale for reaction (4) to proceed in the midstratosphere is of order 5 s. Under normal (preburst) conditions, the timescale for reaction (3) is of order 30 s, so it has only a modest effect. The opposite extreme within our range is when the maximum NO concentration is reached, the highest single concentration region in any place in any of our models would cause the timescale of reaction (3) to drop to 0.2 s. However, a more typical postburst global value for would cause reaction (3) to eventually reach a timescale comparable to reaction (4), of order 5 s. This introduces a rate-limiting, negative feedback effect as N is removed. However, this rate-limiting reaction can only be significant after a great deal of NO is formed, by reaction (4), so it is less effective in our shortest bursts. It would also lose some effectiveness in our very longest duration cases; latitudinal transport over a period of months will move NO around, while N will still be formed primarily at higher altitudes and at latitudes which maximize the irradiation by the source. When the NO and N are not in proximity, reaction (3) will have much more difficulty in moving forward. To summarize, reaction (3) can only compete with reaction (4) for times that are long enough to allow significant NO to build up while the irradiation is on, but not so long that transport can move NO out of the region where the radiation is primarily absorbed. So, for the most part, the effective recombination in reaction (3) plays a minor role.

Ozone depletion timescales in place are fairly rapid, reaching local “equilibrium” values within minutes (but days in the lower stratosphere, where NO\(_x\) concentrations are lower, see discussion in \S\ 3.4). Over a period of weeks reactions, even in the lower stratosphere, take place that maximize ozone destruction with the available catalysts. Therefore, the O\(_3\) depletion rather closely tracks the NO\(_x\) abundance. Some small amount of NO\(_x\) is destroyed by photolysis in the upper stratosphere and mesosphere. The much more important and longer term process for NO\(_x\) destruction is downward transport into the troposphere and rainout as HNO\(_3\), which begins within months and takes nearly a decade to complete. To summarize, most of the NO\(_x\) that is not destroyed in the first few minutes by reaction (3) is likely to survive for months and years.

The long-term history of the global average ozone depletion for various burst durations is shown in Figure 7a, and its behavior on the first day for the shorter bursts is shown in Figure 7b. As discussed previously, the reactions in the lower stratosphere
can take days to roughly equilibrate, given lower NO\textsubscript{3} concentrations there. This effect is of course not visible in those irradiation events, which take place over a timescale longer than this time; in such cases the delay is dominated by the delayed arrival of all the photons. Eventually they all reach approximately the same depletion.

There are of course some limits on the use of the fluence approximation; it is not absolute. There are weak trends with duration. Overall ozone depletion is maximized for very short and very long bursts, with a plateau at 10\textsuperscript{3}–10\textsuperscript{6} s. The trend for very short bursts is associated with the NO destruction timescale. During and soon after a burst, there is considerable atomic nitrogen present. This destroys NO through the reaction (3), as discussed above. Our shortest bursts (0.1 s) begin to deplete ozone as soon as their reaction products are formed (usually beginning with reaction [4]), since many of the photons are captured within the stratosphere and their products do not need to be transported in order to begin damage. Reaction (4) makes NO, so if that has completed, then there is more probability of reaction (3) taking place. Therefore, 10 s bursts have some increased competition with the reaction (3), and at 10\textsuperscript{3} s or longer there is even more competition. Therefore, these bursts deplete slightly less ozone. A plateau is reached. Then, above 10\textsuperscript{3} s transport to the troposphere and rainout just begins to remove NO\textsubscript{3} from the atmosphere, and ozone destruction/recovery timescales come into play (see Thomas et al. 2005b, Figs. 3 and 7). NO\textsubscript{3} compounds created over very extended time periods show a delayed and slightly intensified depletion effect, as evident in Figure 7b. The delay is a straightforward consequence of delayed arrival of photons, and this is the main effect. An NO molecule in a depleted atmosphere has a lower probability of resulting in additional ozone depletion. However, when the atmosphere has begun to recover, the photon has a higher probability of resulting in ozone depletion than if it had arrived earlier. So, events longer than 10\textsuperscript{3} s keep the NO abundance higher at later times, and each NO molecule will have slightly increased effectiveness as compared with it being formed all about the same time in a short burst. On a timescale of longer than a year, vertical mixing can become important through mixing between the stratosphere and the troposphere and transport via the Hadley cells and related phenomena, which cause the 10\textsuperscript{8} s irradiation to show a substantially increased depletion. On even longer scales, rainout will dominate and decrease depletion.

4. DISCUSSION

4.1. Effect of Burst Duration and Temporal Variability

We have found that over many orders of magnitude, holding the total fluence constant, that varying the burst duration over the range of 0.1–10\textsuperscript{8} s had a small effect on the overall atmospheric consequences at the Earth. To first order the ultimate level of ozone depletion was nearly the same, with onset delayed until all the photons had arrived. This is undoubtedly a consequence of the fact that once a photon ionizes and dissociates atmospheric molecules, there are a variety of reaction pathways largely independent of the ionization of other molecules, since the ionization fraction is never close to unity for our range of fluences, and the reaction products usually do not reach high enough abundances to substantially feed back onto their rates. This was anticipated, based on a rough comparison of effects between Thomas et al. (2005b) and Gehrels et al. (2003), which dealt with GRBs and supernovae, respectively. This suggests a simplifying approximation that ozone depletion effects depend primarily on the integrated fluence of an ionizing event. We have seen some increased depletion for timescales at the very short (<10 s) and very long (>10\textsuperscript{3} s) end of our range, but these variations are a few percent, comparable to, for example, day-night differences in the timing of a burst event, and smaller than variations due to latitude or season of the burst. There are strong short-timescale fluctuations in the prompt emission from many GRBs, and strong flares may emerge in their wake (e.g., Falcone et al. 2006). The shock breakout of a supernova in a probable Wolf-Rayet star has shown soft X-ray emission over about 2000 s and UV emission over about a 30 ks timescale (Campana et al. 2006). Cygnus X-3 has been going for years with high-energy photons, at a luminosity that would...
strongly affect any terrestrial-type planet within a few parsecs. This is of course improbable, given the number of objects such as Cygnus X-3 known, but it illustrates the usefulness of these results to estimate the effect of any new candidate objects of interest for terrestrial effects. Our results make possible a simplification in understanding the atmospheric chemistry effects of various kinds of ionizing photons. The ultimate atmospheric effect will depend on the integrated fluence, and the short-timescale variability need not be included in estimating atmospheric effects on terrestrial planets.

Two caveats must be added to this independence claim. While the long-term atmospheric O₃ depletion may be nearly independent of burst duration, some of the biological impact may not be. The major part of the O₃ depletion timescale seen here from a nearly impulsive burst is of order 6 hr (while there may be a more gradual change over the previously mentioned weeks timescale). Thus, while the long-term atmospheric effects are nearly independent of duration, the biological effects may worsen for events exceeding 10⁴ s, since a greater fraction of the UVB present in the event or generated in the upper atmosphere could reach the ground in a partially ozone-depleted atmosphere. Any postburst flares (e.g., Falcone et al. 2006) that occur after this time and contain UVB would impact an already O₃-depleted atmosphere, and more UVB would reach the ground from the burst itself, at least on one side of the planet. For long events such as nearby supernovae, the ozone depletion develops more gradually, and may not reach full intensity for months. This can be more fully understood from the Appendix. The independence from flux is true for the long-term depletion. However, for longer events the depletion at the end of the irradiation period is larger, even if the total fluence is the same. Although, the shorter events with the same fluence will, by the time the longer event has ended, typically have caught up. Aside from this, for a wide range of durations, given an assumed integrated spectrum, one need only know the integrated fluence in order to estimate the long-term atmospheric effects.

4.2. Effect of Spectral Variation

For a peak energy from about 2 keV up to about 20 MeV, the atmospheric perturbation increases (as measured by ozone depletion), with effects nearly constant for even greater peak energy. This reflects the nearly constant attenuation rate of photons above a few MeV in the atmosphere. When our peak is at 20 MeV, a very small proportion of the flux is below 2 MeV. Higher energy photons are attenuated less, and penetrate more deeply before interacting. As seen in the figures, this causes more odd nitrogen production in the stratosphere, where ozone is most abundant. The effect is significant, approximately doubling the DNA-damaging UVB load over our range in peak photon energy.

One might anticipate stronger energy dependence. It does not exist for several reasons. Ozone is present over a range of altitudes, is not a constant fraction of the atmosphere (see Fig. 1), and due to the approximately exponential overall mass density profile of the atmosphere, increased photon energy makes only a moderate change in the penetration depth. O₃ depletion also varies much less than linearly with increased irradiation (as noted in Thomas et al. 2005b). There are three reasons for this. (1) The rate-limiting reaction (3) is more significant if more NO is present. (2) If more ozone is depleted, there is less available to deplete. (3) Higher irradiation may lead to more ozone synthesis, which makes the net depletion less. We are plotting the overall ozone mass, and as seen in Figure 4, for increased photon energy there is increased synthesis of ozone at lower altitudes (by solar UV), and this offsets the depletion effect somewhat. We include it in our overall results, because it is still a part of the relevant optical depth for UVB to reach the surface.

Detailed modeling of a candidate event requires spectral information. One can make some generalizations; softer events such as X-ray flares from the Sun, dominated by low keV photons, are much less damaging for the same fluence than harder events. In particular, SHBs, which have not been included in studies of terrestrial planet effects, may acquire a new significance; they have a harder spectrum than long GRBs, and new data suggest their rate at low redshift is higher than formerly thought (Guetta & Piran 2006). If, as consistent with present data, their energies are 2 orders of magnitude lower and their rate is 2 orders of magnitude larger than LSBs, then they are of competitive expected fluence in an effectively thin disk galaxy. It will be important to use emerging new estimates of the SHB rate and improved spectral information to assess whether they may be competitive with LSBs as an irradiation hazard for terrestrial planetary biospheres. Data from the Gamma-Ray Large Area Space Telescope (GLAST) on the high-energy spectrum will be useful here. Regarding LSBs, described by the Amati et al. (2002) and Amati (2006) relation, the typical peak photon energy of LSBs varies as the square root of the isotropic equivalent energy; therefore, more luminous ones will usually be more damaging than implied by simple luminosity scaling. Early X-ray flares with substantial fluence may also offer an additional hazard if ozone depletion is already underway. Ordinary supernovae may be more damaging than assumed previously, as part of their fluence arrives in an ozone-depleted atmosphere.

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APPENDIX

A HEURISTIC DERIVATION OF THE INDEPENDENCE OF LONG-TERM OZONE DEPLETION FROM BURST DURATION

This derivation simplifies the O₃ depletion to the dominant processes that take place during and after the irradiation period. It is done to illustrate why it is that when the irradiation energy is held constant, but its length varied, that O₃ depletion at the end of the irradiation period increases with its duration, but that at later times the ozone depletion level reached is independent of the duration of the earlier irradiation period. The atmosphere is assumed to be initially dominated by its usual constituents N₂ and O₂. The reactions that dominate each process are included here, except that NO is taken as a proxy for both itself and NO₂ in depletion of O₃, as they are closely coupled in the catalysis. In this notation, [x] refers to the density of x. Rate constants kₛ are as given in Thomas et al. (2005b). The variable J is the
radiation flux onto the atmosphere, and $\sigma$ a typical cross section for ionization and dissociation of $N_2$. Atomic nitrogen formed from dissociation quickly reacts to form NO,

$$d[N]/dt = \sigma J[N_2] - d[NO]/dt,$$

$$d[NO]/dt = k_{16}[O_2][N],$$

therefore,

$$d[N]/dt = \sigma J[N_2] - k_{16}[O_2][N].$$

This is soluble by elementary methods,

$$[N] = (\sigma J[N_2])/(k_{16}[O_2]) - C_1 \exp(-k_{16}[O_2]t),$$

where $C_1$ is a constant reflecting a transient we set equal to zero since N is normally close to nonexistent in the atmosphere, and we have a “steady state” $[N]$ that reflects the ratio of synthesis to loss, and is valid during the burst.

Substituting the resulting value of $[N]$ into reaction (A2), we have

$$d[NO]/dt = \sigma J[N_2],$$

reflecting the fact that NO production is controlled by the flux, assuming that essentially all the N reacts this way, and that NO sinks (particularly rainout) have not yet become important.

Ozone depletion can be described by

$$d[O_3]/dt = -[O_3][NO]k_7.$$  

Substituting $\sigma J[N_2]/t$ from reaction (A5) for $[NO]$ into reaction (A6), we have

$$d[O_3]/dt = -[O_3]k_7\sigma J[N_2]t,$$

implying

$$[O_3] = [O_3]_0 \exp(-0.5k_7\sigma J[N_2]t^2).$$

However, the above is only valid during the period of irradiation, which we call $t_1$. The total fluence is $F = Jt_1$, and we can write the ozone abundance at the end of this period as

$$[O_3]_{t_1} = [O_3]_0 \exp(-0.5k_7\sigma J[N_2]Ft_1).$$

Now, we calculate the continuing ozone depletion with the existing NO, taken as constant before rainout becomes important. From reaction (A5) the NO is frozen at the end of $t_1$ as

$$[NO] = \sigma J[N_2]t_1 = \sigma [N_2]F.$$  

Using this value in reaction (A6) for the postirradiation period, we have

$$d[O_3]/dt = -[O_3]k_7\sigma [N_2]F,$$

which has the solution, using appropriate boundary conditions and evaluated at some later time $t_2 > t_1$,

$$[O_3]_{t_2} = [O_3]_{t_1} \exp\{-k_7\sigma [N_2]F(t_2 - t_1)\}. $$

Substituting reaction (A9) into reaction (A12),

$$[O_3]_{t_2} = [O_3]_0 \exp\{-0.5k_7\sigma [N_2]F(t_2 - 0.5t_1)\}. $$

From reaction (A9) the amount of depletion at the end of the irradiation period $t_1$ increases with $t_1$. This can be verified by a careful examination of Figure 7. For times $t_2 \gg t_1$ the depletion is essentially independent of $t_1$. This condition can be examined in our numerical results and verified for all but our $10^8$ s burst, which is the primary motivation of this derivation. For irradiation times longer than a few years, rainout begins to be important, which is a sink for NO, and the size of effects will obviously be much smaller.
These results can be understood much more simply, given the assumptions made. If basically all N that is produced goes into NO, and NO has no sinks except for rainout (which starts much later), then the amount of NO produced will reflect the total fluence. The mean NO concentration during the irradiation period will be the same for any such period, but for a longer period it will have more time to deplete O₃, which explains the increase in depletion during the burst for longer bursts. As NO is a catalyst that depletes O₃ without being consumed, the ultimate long-term depletion reached will depend on the fluence alone, until rainout begins to be important at very long times.

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