A Self-Immolative Molecular Beacon for Amplified Nucleic Acid Detection

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Fluorogenic hybridization probes allow the detection of RNA and DNA sequences in homogeneous solution. Typically, one target molecule is activating the fluorescence of a single probe molecule. This limits the sensitivity of nucleic acid detection. Herein, we report a self-immolative Molecular Beacon (iMB), which escapes the one-target-one-probe dogma. The iMB probe includes a photoreductively cleavable N-alkylpicolinium (NAP) linkage within the loop region. A fluorophore at the 5'-end serves, on the one hand, as a reporter group and, on the other hand, as a photosensitizer of a NAP-linker cleavage reaction. In the absence of a target, the iMB adopts a hairpin shape. Quencher groups prevent photo-induced cleavage. The iMB opens upon hybridization with target, and both fluorescent emission as well as photo-inductive cleavage of the NAP-linker can occur. In contrast to previous chemical amplification probes, iMBs are unimolecular. Cleavage leads to products that have lower target affinity than the probes before reaction. Aided by catalysis, the method allowed the detection of 5 pM RNA target within 100 min.
A self-immolative molecular beacon for amplified nucleic acid detection

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Abstract: Fluorogenic hybridization probes allow the detection of RNA and DNA sequences in homogeneous solution. Typically, one target molecule is activating the fluorescence of single probe molecule. This limits the sensitivity of nucleic acid detection. Herein, we report a self-immolative Molecular Beacon (iMB), which escapes the one-target-one-probe dogma. The iMB probe includes a photoreductively cleavable N-alkyl-picolinium (NAP) linkage within the loop region. A fluorophore at the 5’-end serves, on the one hand, as a reporter group and, on the other hand, as a photosensitizer of a NAP-linker cleavage reaction. In the absence of target, the iMB adopts a hairpin shape. Quencher groups prevent photo-induced cleavage. The iMB opens upon hybridization with target, and both fluorescent emission as well as photo-reductive cleavage of the NAP-linker can occur. In contrast to previous chemical amplification probes, iMBs are unimolecular. Cleavage leads to products that have lower target affinity than the probes before reaction. Aided by catalysis, the method allowed the detection of 5 pM RNA target within 100 min. Measurements in cell lysate and RNA extract point to the robustness and sequence specificity of the template-triggered photoimmolation

Probe molecules that fluoresce upon recognition of specific nucleic acid targets are invaluable tools for applications in cell biology, biotechnology and molecular diagnostics. Molecular beacons (MBs) (Figure 1A) are a prime example of such turn-on signaling probes.[1] MBs and related probes[2] do not require enzymes and can, therefore, be applied in various settings from in vitro to in vivo scenarios. By design, MB probes adopt a hairpin structure, formed upon intramolecular hybridization of self-complementary stem units. This arranges a quencher in the immediate vicinity of the fluorophore. In the absence of target, the fluorescence is low. Hybridization of a target with the loop segment disrupts the stem, separates the two dyes and restores fluorescence. A host of MB modifications have been introduced with the aim to improve target specificity and the fold fluorescence enhancement.[3] However, without added enzymes a MB probe remains bound to the target and, as a result, the fluorescence enhancement will be low when probes are present in large excess of target; for example, when concentration of the nucleic acid target in a biological sample is low.

Chemical amplification methods provide a solution to the one-target-activates-one-probe issue.[4] In a typical scenario, annealing of two or more reactive oligonucleotide conjugates on the target triggers a proximity-induced chemical reaction that activates fluorescence emission. The existing chemical reaction systems belong to one of three categories; 1) ligation reactions, 2) functional group interconversions or 3) reactions that are catalyzed upon alignment of metal ligands with the aid of a template. Though high sensitivities in the picomolar range have been reported for reactive probes, several hours of reaction are normally required to accumulate signal.[5]
The reported reaction systems typically involve the interplay and optimization of two or even more functionalized oligonucleotides. We envisioned a unimolecular, MB-like probe that allows the sensitive detection of nucleic acids in homogenous phase without added reagents or enzymes. To escape the one-target-activates-one-probe dogma, the unimolecular probe must undergo a cleavage reaction leading to product fragments that have a lower affinity for the target than the probe before reaction. Under these conditions, multiple reactions can proceed on a single target, hence enabling a catalytical chemical signal amplification.

We herein introduce self-immolative molecular beacon (iMB) probes, which include a photocleavable linker unit within the loop-region (X-Y in Figure 1B). The cleavage reaction is catalyzed by a fluorophore such as the coumarin 1 which serves on the one hand as a reporter group and on the
other hand as a photosensitizer of the cleavage reaction. Analogously to conventional MBs, the iMB probe is envisioned to adopt a hairpin shape in absence of target. This positions a quencher in vicinity of the fluorophore. Accordingly, fluorescence and also photosensitizing activity remain low. Target binding separates the fluorophore from the quencher (→ 2). This restores fluorescence and activates the fluorophor’s ability to photosensitize cleavage. The ternary complex formed upon cleavage has lower target affinity than the iMB probe prior cleavage. The product complex 3 can dissociate and the target will be able to act as a catalyst with little interference from product inhibition commonly observed in known templated reactions.

The photocleavable linker is the key unit of the iMB design. Fast photocleavage reactions have been reported for the N-alkyl-picolinium (NAP) protecting group.[6] Falvey showed that [Ru(bpy)3]2+ complexes are efficient photocatalysts of reductive cleavage.[7] Winssinger applied the method for the DNA template-controlled liberation of NAP-caged fluorophores.[8] Falvey also reported liberation of amino acids and phosphates from NAP esters using high wavelength laser dyes acting as photosensitizers.[9] The metal-free photorelease reactions were performed in organic solvents. In test reactions, we dissolved a NAP-protected phenol in an aqueous buffer containing 10 mM ascorbate as reductant and the ATTO425 (1) dye as photosensitizer (Figure S11). Gratifyingly, irradiation with a 455 nm blue LED induced efficient cleavage, also in the aqueous environment. As previously suggested, a likely cleavage mechanism involves a single electron transfer step from ascorbate to the dye’s excited state and from there to the electrophilic NAP group (Figure 1C).[10] The resulting pyridine radical 9 then expells the phenolate leaving group 10. The reduction is complete when the N-alkylpicolinium radical 11 formed upon cleavage abstracts hydrogen from ascorbate or the ascorbate radical.

To introduce the NAP linker in self-immolative molecular beacons (iMB, Figure 1B), we used the bifunctional handle 13 in coupling reactions with two commercially available oligonucleotides (Scheme S2). In this proof-of-concept study, the iMB was designed for recognition of a 20-mer RNA segment transcribed from the b2a2 BCR-ABL fusion gene. A 5-mer stem (60% GC-content) was used to bring the ATTO425 dye into the vicinity of the quencher unit, which was comprised of three BMN535 dyes to allow efficient quenching of fluorescent emission. Fluorescence spectra of the iMB before and after hybridization with an RNA target confirmed that the iMB behaved like a “traditional” MB probe (Figure 2A and Figure S12). Despite the disruption of contiguous base pairing, introduced by the NAP linker, the hybridization with one equivalent RNA target induced an opening of the hairpin, as inferred from the 8-fold increase of fluorescence at 485 nm.

Figure 2. A) Fluorescence spectra of iMB in absence and presence of target (5'-CCAUCAAUAAGGAAGAAGCC-3'). B) Time course of templated photocleavage of iMB. Conditions: 10 mM NaH2PO4, 100 mM NaCl, 2.5 mM MgCl2, 5 mM Asc., pH = 7.4, T = 37 °C; for B): irradiation at λ = 455 nm, yield determined by UPLC-FL (see figure S14A).
Next, we examined the photo-induced self-immolation of the iMB by means of fluorescence detected UPLC analysis. In absence of target, no new product appeared after three hours irradiation time (Figure S14B). By contrast, the addition of one equivalent target to 50 nM iMB triggered a rapid photo cleavage (Figure 2B, Figure S14A). Assuming pseudo-first order kinetics, we determined a reaction half-time \( t_{1/2} = 3.3 \) min (Figure 2B). Importantly, a surprisingly rapid cleavage was also obtained when the reaction was performed in presence of 0.1 equivalents template. After 45 min, cleavage had occurred in 90% yield. This suggests that the products of the photo-induced cleavage reaction are readily displaced by excess iMB probe.

To probe the target affinity of the iMB photo cleavage products we performed UV melting experiments (Figure S17 and S18) with fragments 4a and 5a (Table 1), which contain the characteristic linker features (e.g., DBCO unit, alkyl spacers). The melting temperatures \( T_M \) provided by the fragments, alone or in combination, were 10-20 °C lower than the \( T_M = 56 \) °C of the intact iMB-target duplex. We infer: self-immolation of the iMB does indeed lead to a loss of target affinity. Interestingly, a comparison of the \( T_M \) values of the iMB in absence and presence of target suggested a surprisingly high stability of the iMB hairpin structure (\( T_M (iMB) = 57 \) °C vs. \( T_M (iMB•T) = 56 \) °C). However, compared to the iMB•T system the iMB melting curve (Figure S17) shows a rather shallow sigmoidal transition indicative of low cooperativity. This may facilitate hairpin opening.

### Table 1. Melting temperatures \( T_M \) of complexes formed upon hybridization of target T with the iMB and its fragments.[a]

| Complex     | iMB | iMB•T | 4a•5a•T | 4a•T | 5a•T | T |
|-------------|-----|-------|---------|------|------|---|
| \( T_M / ^\circ \)C | 57  | 56    | 43      | 48   | 38   | 39 |

[a] Conditions: 500 nM oligonucleotide, PBS-Buffer (10 mM NaH2PO4, 100 mM NaCl, 2.5 mM MgCl2, 5 mM Asc., pH = 7.4, detailed structures see Figure 3. F: ATTO425; Q: 3xBMN535. Melting curves see Figures S15 – S18.

Encouraged by the loss of target affinity upon photo-self-immolation, we analyzed fluorescence signalling after irradiation of the iMB in presence of substoichiometric target amounts. With 50 nM iMB, the presence of 5 nM (0.1 eq) target was signalled by an 150% increase of fluorescence at 485 nm (Figure 3). Signalling was fast. A 100% signal increase was obtained after 15 min. Without photoirradiation the non-cleaved iMB afforded a 52% signal increase, which corresponds to the signal expected for ‘traditional’ non-cleavable MB probe with similar fluorescence turn-on characteristics (Table 2). This signal increase afforded by the non-cleaved probe falls to 11% or 4% with 0.02 or 0.005 eq target, respectively. With 125% and 64% change of signal, much higher signalling was observed after 220 min irradiation. Of note, marked 30% signal increases were obtained already after 45 min or less (Figure 3). These experiments point to the advantage of self-
immolating MB probes: a single target molecule can activate the fluorescence of many probe molecules.

![Graph showing time course of fluorescence signalling upon iMB photocleavage at substoichiometric target T.](image)

**Table 2.** Signal enhancement prior to and after photoirradiation of iMB.

| Equivalents T | Signal change[^a] without cleavage / % | Signal change[^a] after 220 min photoirradiation / % |
|--------------|---------------------------------------|-----------------------------------------------|
| 0.1          | 52 ± 3.5                              | 147 ± 2                                       |
| 0.02         | 11 ± 3.1                              | 124 ± 44                                      |
| 0.01         | 7 ± 1.4                               | 100 ± 40                                      |
| 0.005        | 4 ± 2.3                               | 64 ± 8                                        |
| 0.002        | Not detectable                        | 39 ± 5                                        |

[^a] Signal change [%] = (F/F₀ − 1) x 100%. Conditions: see Fig. 3

The analysis of template-induced fluorescence signalling upon photo-cleavage shown in Figure 3 revealed that the maximum signal enhancement (see Figure 2A and Figure S12) was never reached. Control experiments showed that the ATTO425 coumarin dye suffers from photo bleaching (see Figure S13). Though bleaching reduces the achievable signal enhancement, it helps to reduce the background signal in absence of template.

Next, we investigated the limit of detection (LOD). After 100 min photoirradiation in PBS buffer, 100 pM target provided a 27% signal enhancement (Figure 4A) that exceeded the standard error of the measurements by a factor of 3 (3σ criterion). For further decreases of the LOD, Tween20 was added to the buffer. This detergent presumably does not only hinder adsorption or unwanted aggregation but potentially serves also as an anti-fade agent. With this adjustment, 5 pM RNA target afforded a 52±8% signal increase after 100 min (compared to 23±4% without target, see Table S2-S3).

In subsequent experiments, we evaluated the sequence specificity of the iMB probe (Figure 4B). For this purpose, the iMB probe was incubated with a single mismatched RNA-target (1Mm), a target with two mismatched nucleotides (2Mm) and a random target (Mr). In addition, a perfectly complementary DNA-target (MDNA) was added. After 100 min photoirradiation, the signal remained virtually unchanged when the iMB was incubated with the random sequence control or the target
containing two mismatched nucleotides. Incubation with the single mismatched target resulted in a negligible signal change. As expected, the complementary DNA target promoted photo cleavage.

Figure 4. Fluorescence signal change from iMB after A) 15 min / 100 min photoirradiation in presence of 0.1-0.0001 eq target T. Dependence of signalling from iMB on B) target sequence and C) number of gap nucleotides opposite the NAP linkage after 220 min. Conditions: 50 nM iMB, 5 nM (or lower when indicated) target, 10 mM NaH2PO4, 100 mM NaCl, 2.5 mM MgCl2, 5 mM Asc., 0.001% Tween20 (if added), pH = 7.4, 37 °C, irradiation at 455 nm. Signal change = (F/F0 – 1) x 100%.

In further control experiments, the targeted segments were separated by an increasing number of gap nucleotides (Figure 4C). Targets including one or two nucleotide insertions still provided for marked signal enhancements. Signaling gradually decreased as the distance between targeted segments was increased. Apparently, the cleavable linker within the iMB loop portion induces a certain degree of tolerance to nucleotide insertions between the targeted segments. We attribute this behavior to a bivalency effect. The spacer-separated oligonucleotide segments within the iMB loop resemble a spacer-linked bivalent receptor, which recognizes a bivalent ligand (that is the target). As previously reported, bivalent interactions loose strength as the distance between the receptor-ligand pairs increases.\[11\] The highest affinity should be provided by a target that enables coaxial stacking of the iMB loop segments and indeed, the highest fluorescent signal was obtained with a “gapless” target. Despite the tolerance for nucleotide insertions, we wish to note that a target that involves two 10 nt segments separated by less than 6 nucleotides should still be unique.

In conclusion, we have developed a photochemical method for signal amplified nucleic acid detection. In contrast to previously reported chemical methods, the self-immolative molecular beacon (iMB) approach is based on template-controlled cleavage leading to products that have lower template affinity than the probe before reaction. The iMB probe studied provided a 5 pM limit for detection of an RNA target in buffer. First experiments in cell lysate and RNA extract (Figure S20 and S21) indicate the robustness of the method, though the limit of detection is affected in such complex
matrices. Picomolar detection limits have also been reported for reaction systems that induced transfer or cleavage of fluorophores, or removal of fluorescence quencher groups (such as azide, tetrazine, vinyl ether or bromine substituents) from fluorophores. It is instructive to compare the results of our study with previous template-controlled photochemical reactions. Winssinger developed a reaction system involving two oligonucleotides and two PNA strands. Triggered by only 2 pM template, a combination of hybridization chain reactions and [Ru]2+-mediated photocleavage of NAP-quenched PNA-fluorophore conjugates provided for a 4% signal change after a 30 h reaction time. The iMB probe required 100 min to signal 5 pM target by means of 30% fluorescence increase relative to the no template reaction. Mokhir used DNA-photosensitizer conjugates for formation of singlet oxygen to induce the cleavage of 9-alkoxyanthracen-linked fluorophore-DNA conjugates. A 10 pm detection limit was reported for a 30 min reaction involving the concerted action of four dye-labeled oligonucleotide conjugates. Contrary to this set-up and all previously reported templated chemistries our method comprises the use of a single probe. Unimolecular hybridization probes such as molecular beacons have been widely applied. With the current iMB system fluorescence-based nucleic acid detection is limited by a relatively modest fluorescence enhancement (8-fold) upon iMB opening, and photobleaching. Alternative photosensitizer schemes may allow for improvements. Based on data from Falvey et al., an exergonic photo-reductive cleavage of NAP linkers should be feasible when conditions of the Rehm-Weller equation

$$\Delta G_{\text{CT}} = 23.06 \left( E_{\text{Ox}} - E_{\text{Red}} - E_{0.0} + 0.06 \right) < 0$$

are met ($E_{\text{Ox}}$ = oxidation potential of photosensitizer, $E_{\text{Red}}$ = reduction potential of NAP = −1.1 V, $E_{0.0}$ = energy of sensitizer singlet state in eV). Some BODIPY dyes and the 6,8-difluoro-coumarin as well as other dyes such as carbopryonines should be applicable. In analogy to work published from Falvey and Winssinger, iMB probes may also involve a quenched [Ru(bpy)3]2+ dye for photo-catalyzing the cleavage of NAP-caged coumarin linkers.

The data obtained in this study proves the feasibility of nucleic acid cleavage as a new category of DNA/RNA-templated reactions. We showed that a self-immolating option improves the sensitivity of DNA molecular beacons. The concept is probably not restricted to the photo-reductive cleavage of NAP-type linkers. Other photo-triggered cleavage reactions are known. With this and the alternative labeling schemes discussed above, we consider the iMB principle as a new reaction paradigm for DNA/RNA detection chemistries. It seems likely that further improvements allowing fluorescence detection of subpicomolar target are within reach.

**Experimental Section**

Experimental Details can be found in the supporting information.

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Keywords
oligonucleotides • fluorescent probes • Molecular Beacon • photo chemistry • RNA recognition

References

[1] a) S. Tyagi, F. R. Kramer, Nat. Biotechnol. 1996, 14, 303–308; b) J. Zheng, R. Yang, M. Shi, C. Wu, X. Fang, Y. Li, J. Li, W. Tan, Chem. Soc. Rev. 2015, 44, 3036–3055.

[2] a) D. M. Kolpashchikov, Chem. Rev. 2010, 110, 4709–4723; b) A. A. Martí, S. Jockusch, N. Stevens, J. Ju, N. J. Turro, Acc. Chem. Res. 2007, 40, 402–409; c) A. Okamoto, Chem. Soc. Rev. 2011, 40, 5815–5828; d) F. Hövelmann, O. Seitz, Acc. Chem. Res. 2016, 49, 714–723.

[3] a) D. M. Kolpashchikov, Chem. Rev. 2010, 110, 4709–4723; b) A. A. Martí, S. Jockusch, N. Stevens, J. Ju, N. J. Turro, Acc. Chem. Res. 2007, 40, 402–409; c) A. Okamoto, Chem. Soc. Rev. 2011, 40, 5815–5828; d) F. Hövelmann, O. Seitz, Acc. Chem. Res. 2016, 49, 714–723.

[4] a) K. Gorska, N. Winssinger, Angew. Chem. Int. Ed. Engl. 2013, 52, 6820–6843. Angew. Chem., 2013, 125, 6956–6980; b) A. P. Silverman, E. T. Kool, Chem. Rev. 2006, 106, 3775–3789; c) T. N. Grossmann, A. Strohbach, O. Seitz, ChemBiochem 2008, 9, 2185–2192.

[5] a) A. Shibata, T. Uzawa, Y. Nakashima, M. Ito, Y. Nakano, S. Shuto, Y. Ito, H. Abe, J. Am. Chem. Soc. 2013, 135, 14172–14178; b) D. K. Prusty, A. Herrmann, J. Am. Chem. Soc. 2010, 132, 12197–12199; c) T. N. Grossmann, L. Röglin, O. Seitz, Angew. Chem. Int. Ed. Engl. 2007, 46, 5223–5225. Angew. Chem., 2007, 119, 5315–5318; e) Y. Hara, T. Fujii, H. Kashida, K. Sekiguchi, X. Liang, K. Niwa, T. Takase, Y. Yoshida, H. Asanuma, Angew. Chem. Int. Ed. Engl. 2010, 49, 5502–5506. Angew. Chem., 2010, 122, 5634–5638; f) C. Holzhauser, H.-A. Wagenknecht, Angew. Chem. Int. Ed. Engl. 2008, 47, 9555–9559. Angew. Chem. 2008, 120, 9697–9701.

[6] a) C. Sundararajan, D. E. Falvey, J. Org. Chem. 2004, 69, 5547–5554; b) C. Sundararajan, D. E. Falvey, Org. Lett. 2005, 7, 2631–2634; c) C. Sundararajan, D. E. Falvey, Photochem. Photobiol. Sci. 2006, 5, 116–121.

[7] J. B. Borak, D. E. Falvey, J. Org. Chem. 2009, 74, 3894–3899.

[8] a) D. Chang, E. Lindberg, N. Winssinger, J. Am. Chem. Soc. 2017, 139, 1444–1447; b) D. Chang, K. T. Kim, E. Lindberg, N. Winssinger, Bioconjug. Chem. 2018, 29, 158–163.

[9] C. Sundararajan, D. E. Falvey, J. Am. Chem. Soc. 2005, 127, 8000–8001.

[10] J. B. Edson, L. P. Spencer, J. M. Boncella, Org. Lett. 2011, 13, 6156–6159.
[11] N. Dubel, S. Liese, F. Scherz, O. Seitz, Angew. Chem. Int. Ed. Engl. 2019, 58, 907–911. Angew. Chem. 2019, 131, 918–923.

[12] T. N. Grossmann, L. Röglin, O. Seitz, Angew. Chem. Int. Ed. Engl. 2008, 47, 7119–7122. Angew. Chem. 2008, 120, 7228–7231.

[13] a) H. Wu, S. C. Alexander, S. Jin, N. K. Devaraj, J. Am. Chem. Soc. 2016, 138, 11429–11432; b) W. A. Velema, E. T. Kool, J. Am. Chem. Soc. 2017, 139, 5405–5411.

[14] a) D. E. Falvey, C. Sundararajan, Photochem. Photobiol. Sci. 2004, 3, 831–838; b) J. B. Borak, D. E. Falvey, Photochem. Photobiol. Sci. 2010, 9, 854–860; c) J. B. Borak, S. López-Sola, D. E. Falvey, Org. Lett. 2008, 10, 457–460; d) J. B. Borak, D. E. Falvey, J. Org. Chem. 2009, 74, 3894–3899.
A self-immolative molecular beacon for amplified nucleic acid detection

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1 Materials and Methods

1.1 Materials
Modified oligonucleotides (S9 and S11) were purchased from Biomers (Ulm, Germany). Chemicals for the purification of DNA such as acetonitrile (Water content < 30 ppm), isopropanol and ethanol were purchased from Roth (Karlsruhe, Germany). All other chemicals were purchased from Acros, Sigma Aldrich, Fluka, TCI or abcr and were used without further purification. Water was purified with an Astacus Milli-Q Ultra Pure Water Purification System from membraPure GmbH (Bodenheim, Germany).

1.2 Instruments and Methods
Semipreparative HPLC of oligonucleotides was performed on a 1105 HPLC System from Gilson equipped with an UV/Vis-detector (260 nm detection wavelength) and a Waters X-Bridge BEH 130 C18 (10 x 150 mm), 5 µm column at 55 °C. The flow rate was set to 7 mL/min. A binary mixture of A (0.1 m triethylammoniumacetate at pH = 7.3) and B (acetonitrile) was used as mobile phase.

Analytical HPLC was performed on a 1105 HPLC System from Gilson equipped with an UV/Vis-detector (260 nm detection wavelength) and a Waters X-Bridge BEH 130 C18 (4.6 x 250 mm), 5 µm column at 55 °C. The flow rate was set to 1.5 mL/min. A binary mixture of A (0.1 m triethylammoniumacetate at pH = 7.3) and B (acetonitrile) was used as mobile phase.

MALDI-TOF mass spectrometry was performed on a Shimadzu Axima Confidence (positive linear mode). As an excitation source, a nitrogen Laser was used at λ = 337 nm. HPA (0.15 m 2,3,4-trihydroxyacetophenone in MeCN/Water (1:1 [v/v] with 0.5 m diammonium-citrate) was used as a Matrix.

Fluorescence measurements were performed on a Cary Varian Eclipse (Agilent technologies) spectrometer equipped with a peltier block by using either suprasil ultra micro quartz cuvettes (100 µL, d = 1 cm) or suprasil quartz cuvettes (1 mL, d = 1 cm) at either 25 °C or 37 °C. All measurements were performed in degassed phosphate buffer (10 m M NaH2PO4, 100 mM NaCl, pH = 7.4) containing MgCl2, sodium ascorbate and Tween20 as indicated (ex: 430 nm, slit: 5 nm, em: 485 nm, slit: 5 nm). All measurements were done in triplicate.

For UPLC analysis a Waters ACQUITY was equipped with an ACQUITY UPLC TUV detector as well as a fluorescence detector (ex: 430 nm, em: 490 nm). An ACQUITY UPLC BEH oligonucleotide C18 (130 A, 1.7 µm, 2.1 · 50 mm) column was used at a flow rate of 0.5 mL/min at 50 °C. As a mobile phase, a binary mixture of A (0.1 m triethylammoniumacetate at pH = 7.3) and B (acetonitrile) was used.

Photo-induced cleavage was performed in F-96 well plates from PerkinElmer sealed with Greiner EASYseal clear A5596 – 100EA (Sigma-Aldrich) seal-film. The well plate was irradiated with a collimated LED-light placed approx. 15 cm above the plate (455 nm, 1 W: Thorlabs, part. Number M455L2-C2 – www.thorlabs.com) using 75%-85% brightness for set times. The well-plate was cooled/heated with the aid of a custom-made-aluminum-block equipped with a Julabo F250 temperature controller (JULABO GmbH).

UV melting curves were measured at 260 nm using a JASCO spectrometer equipped with a peltier block in UV quartz cuvettes (d = 1 cm), V = 1mL, c = 500 nM. The oligonucleotides were mixed in a 1:1 ratio. The melting curves were recorded at a rate of 0.5 °C/min (20 °C – 90 °C). This was repeated 4x per melting curve. The absorption was normalized and the melting temperature was calculated from the first derivative.

Cell lysate: To release the adherent HEK293WT cells from the surface of the culture dish, the medium was removed and washed once with PBS. Subsequently 0.25% trypsin EDTA (ThermoFischer) was added and incubated for 3-5 min at 37 °C. The cells were detached from the culture dish and the trypsin reaction was stopped by adding full medium (DMEM, 10% FBS (serum), 1x streptomycin/penicillin). The suspension was centrifuged (200 x g, 5 min). The supernatant was discarded. Next, the cells were resuspended in medium and counted (ThermoFischer Countess). The desired number of cells were centrifuged and washed with PBS buffer. The cells were then redissolved in TRIS buffer and proteinase K (3-4 h, 50 °C, 250 µg/mL final conc.) was added. After centrifugation ((200 x g, 5 min, RT), the desired lysate was filtrated before use.

RNA-extract: a cell pellet from 437 mio. HEK293WT cells was dissolved in 15 mL trizol (cooled). Cells were subsequently lysed by pipetting up and down until completely dissolved. Next, 3 mL chloroform was added and the mixture was agitated for 15 sec. The mix was incubated for 3 min at room temperature followed by centrifugation for 15 min at 13.000 rpm (4 °C). The upper layer containing the extracted RNA was carefully pipetted into a sterile falcon tube. One volume-part (pipetted volume of extracted RNA) of cold iPrOH was added along with 1/10-Volume parts of 3 M Na-Acetate. After precipitation overnight, the falcon was centrifuged at 4 °C for 15 min at 4000 rpm. The iPrOH was decanted and the precipitate was washed 10 x with cooled 70% EtOH-solution. The pellet was dried over argon and then resuspended in DEPC-Water. The concentration as well as the purity were determined via Nanodrop: Determined ratios (absorbance maxima): 260/280: 1.98; 260/230: 2.3.
NMR spectra were recorded on a BRUKER Avance II (400 MHz) using CDCl₃, CD₃OD or DMSO-d6 as solvent, and were referenced with respective residual ¹H-solvent peaks of CHCl₃ (δ = 7.26 ppm), CHD₂OD (δ = 3.31 ppm), and DMSO (δ = 2.50 ppm) and ¹³C-solvent peaks of CDCl₃ (δ = 77.0 ppm), CD₃OD (δ = 49.05 ppm), and DMSO-d6 (δ = 39.43 ppm).

LC-MS spectra were recorded using a Waters ACQUITY UPLC coupled with a QDa Mass Spectrometer System (electrospray ionization (ESI)) operating in positive mode on a Waters Acquity UPLC BEH C18 1-7 µm column.

2 Synthesis and Characterization

2.1 Synthesis of the N-Alkyl-picolinium cleavage linker (NAP-CL)

![Scheme S1. Synthesis of the NAP linker](image)

**2.1.1 4-tert.-Butyloxy carbonylamidophenyl-picolyl ether (S₃)**

In a two-neck flask under Ar, 4-N-Boc-aminophenol (S₁) (5.97 g, 28.0 mmol, 1.0 eq) was dissolved in 60 mL dry DMSO. After heating to 50 °C, NaOH (2.80 g, 70 mmol, 2.5 eq) was added. The suspension was stirred at this temperature for 1 h. Next, 3-chloro-methylpyridine (S₂) (4.64 g, 28.0 mmol, 1.0 eq) was added within 2 h. The reaction mixture was heated to 80 °C and stirred for 18 h. After cooling to room temperature, the mixture was poured onto 500 mL of ice and extracted from EtOAc (10 x 300 mL). The combined organic layers were washed with water (10 x 300 mL) and dried over MgSO₄. The solvent was removed in vacuo and the crude product was purified via flash-chromatography (DCM/EtOAc 4:1) to yield S₃ (4.93 g, 60%) as a white solid.

**1H-NMR** (500 MHz, CDCl₃): δ [ppm] = 8.60 (dd, J = 4.5, 1.5 Hz, 2H), 7.35 (d, J = 6.1 Hz, 2H), 7.28 (d, J = 8.3 Hz, 2H), 6.89 – 6.87 (m, 2H), 6.51 (s, 1H), 5.05 (s, 2H), 1.50 (s, 9H).

**13C-NMR** (126 MHz, CDCl₃): δ [ppm] = 154.2, 153.2, 149.9, 146.6, 132.4, 121.1, 120.6, 115.3, 68.6, 28.5. LC-MS (ESI) RT = 2.4 min (3-80% B in 4 min). m/z calcd for C₁₇H₂₀N₂O₃ +H+: 301.15 [M+H]+, found: 301.28 [M+H]+.

3-Azido-1-propanol (S₅)[¹]

In a 100 mL flask, 3-Chloro-1-propanol (S₄) (1.89 g, 20.0 mmol, 1.0 eq) was dissolved in 40 mL ddH₂O. Sodium azide (2.60 g, 40.0 mmol, 2.0 eq) was added successively. The reaction mixture was heated to 80 °C and stirred for 72 h. After cooling to RT, the crude mixture was extracted with DCM (4 x 70 mL). The combined organic layers were dried over MgSO₄. The solvent was
removed in vacuo (35 °C, max. 150 mbar) to yield the product (S5, 2.02 g, 99%) without further purification as a clear colorless oil. The NMR-data is in agreement with the given literature-data\[1]. 1H-NMR (500 MHz, CDCl3): δ[ppm] = 3.73 (t, J = 6.0 Hz, 2H), 3.44 (t, J = 6.6 Hz, 2H), 1.98 (s, 1H), 1.91 – 1.72 (m, 2H). 13C-NMR (126 MHz, CDCl3): δ[ppm] = 60.0, 48.5, 31.2.

2.1.2 3-Azido-1-triflyl-propanol (S6)\[2]

In a 100 mL flask under Ar, 3-Azido-1-propanol (S5) (2.00 g, 20.0 mmol, 1.0 eq) was dissolved in 45 mL DCM and cooled to 0 °C. At this temperature, pyridine (2.62 g, 24.0 mmol, 1.2 eq) was added and subsequently cooled to –78 °C. Next, Triflic-anhydride (6.23 g, 22.0 mmol, 1.1 eq) was added within 20 min. The reaction mixture was stirred for another 3 h at –78 °C and then warmed to RT. 30 mL sat. NH4Cl was added to quench the reaction followed by an extraction with EtOAc (3 x 70 mL). The combined organic layers were dried over MgSO4. The solvent was removed in vacuo and the crude was purified with flash-chromatography (Hexane/EtOAc 10:1) to yield the product (S6, 3.91 g, 84%) as a clear colorless oil. The NMR-Data is in agreement with the given literature-data\[2].

1H-NMR (500 MHz, CDCl3): δ[ppm] = 4.63 (t, J = 6.0 Hz, 2H), 3.52 (t, J = 6.4 Hz, 2H), 2.08 – 2.06 (m, 2H).

13C-NMR (126 MHz, CDCl3): δ[ppm] = 73.9, 46.9, 28.9.

2.1.3 4-tert.-Butyloxycarbonylamidophenyl N-azidopropylpicolinium ether (S7)

In a 250 mL two-neck flask under Ar, S3 (600 mg, 2.0 mmol, 1.0 eq) was dissolved in 170 mL dry DCM. The solution was cooled to –78 °C and 3-azido-1-triflyl-propanol (S6) (653 mg, 2.80 mmol, 1.4 eq) was added successively. The reaction mixture was warmed to RT and stirred for another 20 h. The solvent was removed in vacuo and the crude product was purified via flash-chromatography (DCM/MeOH 10:1) to yield S7 (1.05 g, 99%) as a white solid.

1H-NMR (500 MHz, CD3OD): δ[ppm] = 8.97 (d, J = 6.8 Hz, 2H), 8.19 (d, J = 6.7 Hz, 2H), 7.37 (d, J = 8.9 Hz, 2H), 7.11 – 6.76 (m, 2H), 5.45 (s, 2H), 4.73 (t, J = 7.2 Hz, 2H), 3.55 (s, 2H), 3.37 (s, 2H), 2.43 – 2.14 (m, 2H), 1.53 (s, 9H).

13C-NMR (126 MHz, CD3OD): δ[ppm] = 160.4, 155.6, 154.5, 145.9, 134.9, 126.5, 123.1, 121.8, 120.5, 116.2, 80.8, 68.6, 60.11, 31.2, 28.7. LC-MS (ESI) RT = 2.7 min (3-80% B in 4 min). m/z calcd for C21F3H26N5O6S: 533.15, found: 384.24 [M-OTf]+.

2.1.4 NAP-CL (13)

To a 50 mL flask under Ar, S7 (1.05 g, 1.98 mmol, 1.0 eq) was added and dissolved in DCM/TFA (1:1, v/v) 16 mL) and stirred for 1 h at RT. The deprotection was followed via LC-MS. After quantitative deprotection, the solvent mixture was evaporated by means of an argon stream, washed 5 x with DCM and dried under reduced pressure for 30 min. The residue was dissolved in water (10 mL) and sat. NaHCO3 was added until the solution reached pH = 8.0 and a white solid precipitated. The precipitate was dissolved in EtOAc followed by an extraction with EtOAc (5 x 60 mL). The combined organic layers were dried over MgSO4 and the solvent was reduced in vacuo. The aniline derivative was used without further purification.

To a 250 mL flask under Ar, aniline intermediate (745 mg, 1.72 mmol, 1.0 eq)  was added and dissolved in dry EtOAc (90 mL). Next, glutaric anhydride (255 mg, 2.24 mmol, 1.3 eq) was added. The reaction mixture was heated to 50 °C until the suspension was completely dissolved. The solution was then stirred at RT for 18 h. The formed precipitate was filtered, washed with cold EtOAc and subsequently dried at reduced pressure. The product (13, 800 mg, 85%) was obtained as a pale-yellow solid and used without further purification. 1H-NMR (500 MHz, CD2OD): δ[ppm] = 8.96 (d, J = 6.8 Hz, 2H), 8.17 (d, J = 6.6 Hz, 2H), 7.59 – 7.39 (m, 2H), 7.09 – 6.92 (m, 2H), 5.45 (s, 2H), 4.71 (t, J = 7.2 Hz, 2H), 3.51 (t, J = 6.3 Hz, 2H), 2.40 (dt, J = 12.7, 7.3 Hz, 4H), 2.33 – 2.17 (s, 2H), 1.98 (dd, J = 16.0, 8.6 Hz, 3H), 1.24 (t, J = 7.1 Hz, 1H). 13C-NMR (126 MHz, CD2OD): δ[ppm] = 179.9, 173.6, 160.2, 155.5, 145.9, 134.3, 126.5, 123.2, 116.1, 68.6, 60.1, 36.8, 34.2, 31.2, 22.2, 20.9. LC-MS (ESI) RT = 1.8 min (3-80% B in 4 min). m/z calcd for C21F3H26N5O6S: 547.13 g/mol, found: 398.38 [M-OTf]⁺.
2.2 NMR-Spectra

Figure S1. $^1$H-NMR (500 MHz, CDCl$_3$) of S5.

Figure S2. $^{13}$C-NMR (126 MHz, CDCl$_3$) of S5.
Figure S3. $^1\text{H-NMR}$ (500 MHz, CDCl$_3$) of S6.

Figure S4. $^{13}\text{C-NMR}$ (126 MHz, CDCl$_3$) of S6.
Figure S5. $^1$H-NMR (500 MHz, CDCl$_3$) of S3.

Figure S6. $^{13}$C-NMR (126 MHz, CDCl$_3$) of S3.
Figure S7. $^1$H-NMR (500 MHz, MeOD-d4) of S7.

Figure S8. $^{13}$C-NMR (126 MHz, MeOD-d4) of S7.
Figure S9. $^1$H-NMR (500 MHz, MeOD-d4) of 13.

Figure S10. $^{13}$C-NMR (126 MHz, MeOD-d4) of 13.
2.3 Photolysis of Linker 13 with ATTO425

**Figure S11.** Photo triggered cleavage of 13. A) UPLC-trace before irradiation and B) close-up after 30, 60 and 120 min; C) calculated yields (integration of peaks). Conditions: 500 µM 13, 2.5 mM ATTO425 in buffer (10 mM NaH₂PO₄, 100 mM NaCl, 2.5 mM MgCl₂, 10 mM Na-Asc., pH = 7.4), irradiation at 455 nm.

2.4 Turn-on (50 nM iMB)

**Figure S12.** Fluorescence spectra of iMB before (dashed) and after (solid) addition of target T. Conditions: 50 nM iMB, 50 nM T (if added) in PBS buffer (10 mM NaH₂PO₄, 100 mM NaCl, 2.5 mM MgCl₂, 5 mM Asc., pH = 7.4), T = 37 °C.

2.5 Photobleaching of ATTO425

**Figure S13.** A) Photobleaching of ATTO425 (50 nM) in PBS buffer (10 mM NaH₂PO₄, 100 mM NaCl, 2.5 mM MgCl₂, pH = 7.4) upon irradiation at 455 nm for indicated time points. Read out in 1000 µL Quartz cuvettes.
2.6 iMB Synthesis

**Scheme S2.** Synthesis of iMB S12

### 2.6.1 S10

A mixture of HATU (10 µmol, 100 eq), DIPEA (30 µmol, 300 eq) and NAP-CL 13 (10 µmol, 100 eq) in DMF (100 µL) was incubated for 4 min. Subsequently, the reaction mixture was transferred to a 2 mL Sarstedt microtube containing oligonucleotide S9 (100 nmol, 1 eq) in 200 µL water. The mixture was agitated for 3 h at 25 °C. The oligonucleotide was precipitated with iPrOH, vortexed and left for quantitative precipitation at –18 °C for 3 h. After centrifugation (10 min, 13000 rpm), the precipitate was dissolved in 100 µL water. Next, 3M sodium acetate (30 µL) was added and the crude product was precipitated with iPrOH. After another cooling-period at –18 °C for 3 h the crude was centrifuged and dried over Argon. The oligonucleotide was purified via HPLC. The combined fractions were lyophilized and the residue was desalted via precipitation with ammonium acetate and iPrOH.

**ATTO425-CTC CAG GCT TCT TCC-ArO-NAP-N3 (S10)**

- **HPLC:** $t_R = 17$ min (10 – 45% B in 25 min)
- **UPLC:** $t_R = 2.28$ min (3 – 60% B in 4 min)
- **MALDI-TOF (m/z) calcd for:** $C_{207}H_{275}N_{58}O_{111}P_{16}$: 5845.35 [M+H]+, found: 5845.0 [M+H]+, 5671.9 [M+H]+ (OH- Fragment). Fragmentation of coupled NAP-CL 13 occurs during MALDI-TOF measurement (laser induced).

### 2.6.2 SPAAC (iMB synthesis)

A 2 mL Sarstedt microtube was charged with a solution of S10 (20 nmol, 1 eq) in a water/ACN (1:1 [v/v], 50 µL) mixture. A solution of DBCO-oligonucleotide (5a) (20 nmol, 1.0 – 1.8 eq) in a water/ACN (1:1 [v/v], 50 µL) mixture was added. The solution was agitated for 72 h at 30 °C. Oligonucleotides were precipitated with iPrOH, vortexed and left for quantitative precipitation at –18 °C for 3 h. After centrifugation (10 min, 13000 rpm), the precipitate was dissolved in 100 µL water. Next, 3M sodium acetate (30 µL) was added and the crude product was precipitated with iPrOH. After another cooling-period at –18 °C for 3 h the crude was centrifuged and dried over Argon. The oligonucleotide was purified via HPLC. The combined fractions were lyophilized and the residue was desalted via precipitation with ammonium-acetate and iPrOH.

**ATTO425-CTC CAG GCT TCT-TCc3-ArO-NAP-Triazol-TTA TTG ATG GTG GAC-3xBMN535 (S12)**

- **HPLC:** $t_R = 17$ min (25 – 50% B in 25 min, 50% B 15 min, 50 – 75% B in 3 min)
UPLC: $t_R = 2.72$ min (3 – 80% B in 4 min)
MALDI-TOF ($m/z$) calcd: 12436.2 [M+H]$^+$, found: 12436.3 [M+H]$^+$, 5673.4 [M+H]$^+$ (OH-Fragment). Fragmentation of iMB S12 occurs during MALDI-TOF measurement (laser induced).

2.6.3 Target-Sequences

Table S1. Sequences of the examined targets. Mismatched nucleobases are highlighted.

| Name | Sequence (5' – 3') |
|------|-------------------|
| T    | CCA UCA AUA AGG AAG CC |
| M_{Dna} | CCA TCA ATA AGG AAG CC |
| 1Mm  | CCA UCA AUA AGG CAG AAG CC |
| 2Mm  | CCA GCA AUA AGG CAG AAG CC |
| Mr   | UGC AUC UUG AAA UGU AAU UCG |
| M_{ox} | CCA UCA AUA AGG GAA GAA GCC |
| M_{ox} | CCA UCA ALA AGC GGA AGA AGC |
| M_{ox} | CCA UCA AUA AGC CGG AAG AAG CC |
| M_{ox} | CCA UCA AUA AGC UCG GAA GAA GCC |
| M_{ox} | CCA UCA AUA AGC UCA GGA AGA AGC |
| M_{ox} | CCA UCA AUA AGC UCA CGG AAG AAG CC |
| M_{ox} | CCA UCA AUA AGC UCA CAG GAA GAA GCC |

2.7 Determination of cleavage yield

Figure S14. Exemplary UPLC-traces of iMB after photo irradiation in A) presence of 0.1 eq. target T or B) in absence of target. Conditions: 50 nM iMB in PBS buffer (10 mM NaH$_2$PO$_4$, 100 mM NaCl, 2.5 mM MgCl$_2$, 5 mM Asc. pH = 7.4), T = 37 °C, irradiation at 455 nm. Traces in A) show the growth of the peak at $t_R = 2.7$ min. This is not observed for traces in B). The rather low signal intensity over time in B) may be due to adsorption.
2.8 Melting ($T_m$) analysis

**Figure S15.** Normalized melting curves of A) IMB alone and B) after addition of 1 eq $T$.

**Figure S16.** Melting curves of A) IMB alone and B) after addition of 1 eq $T$.

**Figure S17.** Normalized melting curves of A) $T$ and B) $5a \cdot T$.

**Figure S18.** Normalized melting curves of A) $4a \cdot T$ and B) $5a \cdot T \cdot 4a$. 
## 2.9 Measurements for limit of detection (LOD)

**Table S2.** Fluorescence intensities, $F/F_0$ and signal change (%) upon photo irradiation of the iMB in presence of varied amounts of target T. Conditions: 50 nM iMB in PBS buffer (10 mM NaH$_2$PO$_4$, 100 mM NaCl, 5 mM Asc., 2.5 mM MgCl$_2$, pH = 7.4), T = 37 °C, irradiation at 455 nm.

| eq (T) | 0 min | 0 min + T | 15 min | 45 min | 100 min | 220 min |
|--------|-------|-----------|--------|--------|---------|---------|
|        | FL / a.u. | FL / a.u. | F/F$_0$ | FL / a.u. | F/F$_0$ | FL / a.u. | F/F$_0$ | FL / a.u. | F/F$_0$ | FL / a.u. | F/F$_0$ |
| 0      | 3.10 ± 0.06 | 3.00 | 0.97 | 2.58 | 0.83 | 2.50 | 0.81 | 2.51 | 0.81 | 2.50 | 0.81 |
|        | 3.22 | 3.22 | 1.00 | 2.60 | 0.81 | 2.71 | 0.84 | 2.84 | 0.88 | 2.80 | 0.87 |
|        | 3.20 | 3.30 | 1.03 | 2.60 | 0.81 | 2.66 | 0.83 | 2.78 | 0.87 | 2.60 | 0.81 |
| **avg** | 3.17 ± 0.16 | 3.17 | 1.00 | 2.59 | 0.82 | 2.62 | 0.82 | 2.71 | 0.85 | 2.63 | 0.83 |
| 0.002  | 3.50 | 3.50 | 1.00 | 3.58 | 1.02 | 3.90 | 1.11 | 4.60 | 1.31 | 5.00 | 1.43 |
|        | 3.50 | 3.59 | 1.03 | 3.59 | 1.03 | 3.79 | 1.08 | 4.20 | 1.20 | 4.96 | 1.42 |
|        | 3.52 | 3.52 | 1.00 | 3.60 | 1.02 | 3.80 | 1.08 | 4.50 | 1.28 | 4.69 | 1.33 |
| **avg** | 3.51 ± 0.05 | 3.54 | 1.01 | 3.59 | 1.02 | 3.83 | 1.09 | 4.43 | 1.26 | 4.88 | 1.97 |
| 0.005  | 4.20 | 4.35 | 1.04 | 5.20 | 1.24 | 5.70 | 1.36 | 6.20 | 1.48 | 6.50 | 1.55 |
|        | 4.02 | 4.30 | 1.07 | 5.00 | 1.24 | 5.70 | 1.42 | 6.40 | 1.59 | 6.70 | 1.67 |
|        | 3.70 | 3.80 | 1.03 | 4.30 | 1.16 | 4.97 | 1.34 | 5.40 | 1.46 | 6.30 | 1.70 |
| **avg** | 3.97 ± 0.25 | 4.15 | 1.04 | 4.83 | 1.21 | 5.46 | 1.37 | 6.00 | 1.51 | 6.50 | 1.64 |
| 0.01   | 3.35 | 3.55 | 1.06 | 4.92 | 1.47 | 5.90 | 1.76 | 6.24 | 1.86 | 7.40 | 2.21 |
|        | 3.42 | 3.68 | 1.08 | 4.59 | 1.34 | 5.34 | 1.56 | 6.05 | 1.77 | 7.69 | 2.25 |
|        | 3.40 | 3.70 | 1.09 | 4.14 | 1.22 | 4.90 | 1.44 | 5.22 | 1.54 | 5.20 | 1.54 |
| **avg** | 3.39 ± 0.04 | 3.64 | 1.07 | 4.55 | 1.34 | 5.38 | 1.59 | 5.84 | 1.72 | 6.76 | 1.72 |
| 0.02   | 3.10 | 3.56 | 1.15 | 5.00 | 1.61 | 6.15 | 1.98 | 6.58 | 2.12 | 7.40 | 2.39 |
|        | 3.47 | 3.82 | 1.10 | 4.87 | 1.40 | 5.94 | 1.71 | 6.89 | 1.99 | 9.00 | 2.59 |
|        | 3.10 | 3.38 | 1.09 | 4.30 | 1.39 | 5.00 | 1.61 | 5.38 | 1.74 | 5.40 | 1.74 |
| **avg** | 3.22 ± 0.21 | 3.59 | 1.11 | 4.72 | 1.47 | 5.67 | 1.77 | 6.28 | 1.95 | 7.27 | 2.24 |
| 0.1    | 4.10 | 6.37 | 1.55 | 8.20 | 2.00 | 10.17 | 2.48 | 9.66 | 2.36 | 10.02 | 2.44 |
|        | 3.37 | 5.00 | 1.48 | 6.90 | 2.05 | 8.40 | 2.49 | 8.70 | 2.58 | 8.40 | 2.49 |
|        | 3.80 | 5.77 | 1.52 | 7.51 | 1.98 | 9.31 | 2.45 | 9.38 | 2.47 | 9.40 | 2.47 |
| **avg** | 3.76 ± 0.37 | 5.71 | 1.52 | 7.54 | 2.01 | 9.25 | 2.47 | 9.25 | 2.47 | 9.27 | 2.47 |
| 1.0    | 4.50 | 25.70 | 5.71 | 28.13 | 5.84 | 5.30 | 4.30 | 5.94 | 28.13 | 5.84 |
Table S3. Fluorescence intensities, F/F₀ and signal change (%) upon photo irradiation of the iMB in presence of varied amounts of target T. Conditions: 50 nM iMB in PBS buffer (10 mM NaH₂PO₄, 100 mM NaCl, 5 mM Asc., 2.5 mM MgCl₂, 0.001% Tween20, pH = 7.4), T = 37 °C, irradiation at 455 nm.

| eq (T) | 0 min | 0 min + T | 15 min | 45 min | 100 min | 220 min |
|--------|-------|-----------|--------|--------|---------|---------|
|        | F/I | F/I | F/F₀ | F/F₀ | F/F₀ | F/F₀ | F/F₀ | F/F₀ |
| 0      | I 3.66 | 3.66 | 1.00 | 3.56 | 0.97 | 4.10 | 1.12 | 4.48 | 1.22 | 5.90 | 1.61 |
|        | II 3.70 | 3.70 | 1.00 | 3.60 | 0.97 | 4.61 | 1.24 | 4.68 | 1.26 | 5.10 | 1.38 |
|        | III 3.70 | 3.70 | 1.00 | 3.00 | 0.81 | 3.70 | 1.00 | 4.40 | 1.19 | 5.20 | 1.41 |
| avg    | 3.69 ± 0.02 | 3.69 ± 0.02 | 1.00 ± 0.04 | 3.39 ± 0.34 | 0.92 ± 0.09 | 4.13 ± 0.45 | 1.12 ± 0.14 | 4.52 ± 0.04 | 1.22 ± 0.04 | 5.40 ± 0.44 | 1.47 ± 0.13 |
| 0.0001 | I 3.50 | 3.50 | 1.00 | 3.80 | 1.09 | 4.60 | 1.31 | 5.60 | 1.60 | 6.00 | 1.71 |
|        | II 3.81 | 3.81 | 1.00 | 4.08 | 1.07 | 4.77 | 1.26 | 5.50 | 1.45 | 6.00 | 1.58 |
|        | III 3.80 | 3.75 | 0.99 | 3.85 | 1.01 | 4.70 | 1.24 | 5.70 | 1.50 | 6.70 | 1.76 |
| avg    | 3.70 ± 0.17 | 3.68 ± 0.16 | 1.00 ± 0.21 | 3.91 ± 0.15 | 1.06 ± 0.04 | 4.69 ± 0.09 | 1.27 ± 0.04 | 5.60 ± 0.10 | 1.52 ± 0.08 | 6.23 ± 0.40 | 1.69 ± 0.10 |
| 0.0002 | I 3.50 | 3.50 | 1.00 | 3.80 | 1.09 | 4.81 | 1.37 | 5.60 | 1.60 | 6.00 | 1.71 |
|        | II 3.80 | 3.85 | 1.01 | 4.13 | 1.09 | 4.92 | 1.30 | 5.70 | 1.50 | 6.20 | 1.63 |
|        | III 3.78 | 3.81 | 1.01 | 4.20 | 1.12 | 5.10 | 1.36 | 6.01 | 1.60 | 6.70 | 1.78 |
| avg    | 3.69 ± 0.17 | 3.72 ± 0.16 | 1.01 ± 0.21 | 4.04 ± 0.21 | 1.10 ± 0.15 | 4.94 ± 0.04 | 1.34 ± 0.04 | 5.77 ± 0.21 | 1.57 ± 0.06 | 6.30 ± 0.36 | 1.71 ± 0.08 |
| 0.0005 | I 3.50 | 3.50 | 1.00 | 4.07 | 1.16 | 4.76 | 1.36 | 5.90 | 1.69 | 6.20 | 1.77 |
|        | II 4.04 | 4.04 | 1.00 | 4.60 | 1.14 | 5.77 | 1.43 | 6.89 | 1.71 | 6.90 | 1.71 |
|        | III 3.80 | 3.80 | 1.00 | 4.30 | 1.13 | 5.30 | 1.39 | 6.20 | 1.63 | 7.20 | 1.89 |
| avg    | 3.78 ± 0.27 | 3.78 ± 0.27 | 1.00 ± 0.27 | 4.32 ± 0.27 | 1.14 ± 0.21 | 5.28 ± 0.04 | 1.39 ± 0.04 | 6.33 ± 0.21 | 1.67 ± 0.06 | 6.77 ± 0.36 | 1.79 ± 0.09 |
| 0.001  | I 3.60 | 3.69 | 1.03 | 4.20 | 1.17 | 5.24 | 1.46 | 6.67 | 1.85 | 7.40 | 2.06 |
|        | II 4.20 | 4.13 | 0.98 | 4.90 | 1.17 | 6.17 | 1.47 | 7.89 | 1.88 | 8.20 | 1.95 |
|        | III 3.87 | 4.10 | 1.06 | 4.61 | 1.19 | 5.78 | 1.49 | 6.70 | 1.73 | 7.37 | 1.90 |
| avg    | 3.89 ± 0.30 | 3.97 ± 0.25 | 1.02 ± 0.35 | 4.57 ± 0.35 | 1.17 ± 0.01 | 5.73 ± 0.47 | 1.47 ± 0.02 | 7.09 ± 0.70 | 1.82 ± 0.08 | 7.66 ± 0.47 | 1.97 ± 0.08 |
2.10 Specificity of iMB signalling

Table S4. Fluorescence intensities, F/F₀ and signal change (%) afforded by iMB prior to and 15 or 100 min after photoirradiation in presence of mismatched RNA. Conditions: 50 nM iMB, 5 nM RNA mix (2Mm, Mr, M₇nt), PBS buffer (10 mM NaH₂PO₄, 100 mM NaCl, 2.5 mM MgCl₂, 5 mM Asc. pH = 7.4), T = 37 °C, irradiation at 455 nm.

| eq (T) | 0 min | 0 min + T | 15 min | 100 min |
|--------|-------|-----------|--------|---------|
|        | Fl / a.u. | Fl / a.u. | F/F₀ | Signal change / % | Fl / a.u. | F/F₀ | Signal change / % | Fl / a.u. | F/F₀ | Signal change / % |
| 0      | 6.20  | 6.40  | 1.03  | 3.23  | 5.78  | 0.93 | -6.77  | 6.65  | 1.07 | 7.26  |
| II     | 6.50  | 6.20  | 0.95  | -4.61 | 5.30  | 0.81 | -18.46 | 6.20  | 0.86 | -4.62 |
| III    | 5.80  | 6.00  | 1.03  | 3.44  | 5.70  | 0.87 | -12.59 | 6.60  | 0.98 | 13.79 |
|        | 6.12 ± 0.35 | 6.20 ± 0.20 | 0.67 ± 4.59 | 5.38 ± 0.36 | 0.87 ± 5.84 | -12.61 ± 0.67 | 6.48 ± 0.25 | 0.96 ± 3.03 | 5.48 ± 0.96 |
| 0.0025 | I     | 6.10  | 6.58  | 1.08  | 7.87  | 6.12 | 1.00  | 6.85  | 1.12 | 12.30 |
| II     | 5.78  | 5.78  | 1.00  | 0.95  | 6.08  | 1.05 | 5.19  | 7.90  | 1.37 | 36.68 |
| III    | 5.60  | 5.70  | 1.02  | 1.79  | 5.80  | 1.04 | 3.57  | 6.18  | 1.10 | 10.36 |
|        | 5.83 ± 0.25 | 6.02 ± 0.49 | 1.03 ± 4.13 | 6.00 ± 0.17 | 1.03 ± 3.22 | 6.98 ± 0.02 | 1.20 ± 3.03 | 19.78 ± 0.15 | 9.33 ± 14.67 |
| 0.005  | I     | 5.35  | 5.90  | 1.10  | 10.28 | 6.24  | 1.17 | 16.64 | 7.72  | 1.44 | 44.30 |
| II     | 5.87  | 6.10  | 1.04  | 3.92  | 6.70  | 1.14 | 14.14 | 7.80  | 1.33 | 32.88 |
| III    | 6.10  | 6.70  | 1.10  | 9.84  | 6.70  | 1.10 | 9.84  | 7.00  | 1.15 | 14.75 |
|        | 5.77 ± 0.38 | 6.23 ± 0.42 | 1.08 ± 3.55 | 6.55 ± 0.27 | 1.14 ± 3.44 | 13.54 ± 0.44 | 1.54 ± 3.06 | 14.63 ± 0.08 | 7.54 ± 16.54 |
| 0.01   | I     | 5.88  | 6.88  | 1.17  | 17.01 | 8.88  | 1.51 | 51.02 | 9.20  | 1.56 | 56.46 |
| II     | 6.59  | 7.70  | 1.17  | 16.84 | 9.60  | 1.46 | 51.90 | 10.13 | 1.54 | 53.72 |
| III    | 6.49  | 7.50  | 1.16  | 15.56 | 9.78  | 1.51 | 57.94 | 10.20 | 1.57 | 57.16 |
|        | 6.32 ± 0.38 | 7.36 ± 0.43 | 1.16 ± 3.55 | 9.42 ± 0.48 | 1.49 ± 3.27 | 54.56 ± 0.02 | 1.56 ± 3.06 | 55.78 ± 1.82 | 7.54 ± 16.54 |
| 0.1    | I     | 6.50  | 9.01  | 1.39  | 38.62 | 11.45 | 1.76 | 76.15 | 14.80 | 2.28 | 127.69 |
| II     | 6.80  | 9.30  | 1.37  | 36.76 | 11.70 | 1.72 | 72.06 | 15.38 | 2.26 | 126.18 |
| III    | 6.24  | 9.10  | 1.46  | 45.83 | 11.10 | 1.76 | 77.88 | 15.50 | 2.48 | 148.40 |
|        | 6.50 ± 0.38 | 9.14 ± 0.43 | 1.40 ± 4.79 | 11.41 ± 0.30 | 1.75 ± 3.00 | 75.36 ± 0.37 | 15.23 ± 0.12 | 134.09 ± 12.41 | 12.41 ± 18.21 |

Figure S19. Signal change afforded by iMB upon photo irradiation in presence of mismatched RNA. Graphical representation of data in Table S4.
2.11 Measurements in RNA-extract and cell Lysate (HEK293WT)

**Figure S20.** Signalling from the iMB in RNA extract. Conditions: 100 nM iMB, PBS buffer (10 mM NaH2PO4, 100 mM NaCl, 5 mM Asc., 0.001% Tween20, pH = 7.4), T = 37 °C, irradiation at 455 nm. The reaction was carried out in 96 well plates. V<sub>react</sub>: 150 µL, read-out: 100 µL suprasil ultra-micro quartz cuvettes.

**Figure S21.** Signal change upon photo irradiation (15 min) of iMB in A) cell lysate or B) RNA extract spiked with indicated equivalents of target T. The blue overlay indicates the theoretical signal change provided by a non-cleavable MB probe with similar fluorescence turn-on characteristics. Conditions: A) 50 nM iMB in 8 vol% lysate from 10 mio. HEK293 in PBS buffer containing 5 mM MgCl₂, 5 mM Asc., pH = 7.4, 37 °C. Read out in 1000 µL suprasil quartz cuvettes. B) 100 nM iMB in RNA extract (5 µg in PBS buffer, 5 mM Asc., 0.001% Tween20, pH = 7.4) at 37 °C. Read-out in 100 µL suprasil ultra-micro quartz cuvettes.

**Table S5.** Measurements in cell lysate. Fluorescence Intensities (a.u.); c (iMB) = 50 nM, PBS-buffer (10 mM NaH2PO4, 100 mM NaCl, 5 mM Asc., 5 mM MgCl₂, pH = 7.4), T = 37 °C, irradiation at 455 nm.

| eq (T) | 0 min | 0 min + T | 15 min | 45 min | 100 min |
|--------|-------|-----------|--------|--------|---------|
|        | FI / a.u. | FI / a.u. | FI/F<sub>0</sub> | FI / a.u. | FI/F<sub>0</sub> | FI / a.u. | FI/F<sub>0</sub> | FI / a.u. | FI/F<sub>0</sub> |
| 0      |        |           |         |        |         |        |           |         |        |
| I      | 4.72   | 4.57      | 0.97    | 3.84   | 0.81    | 3.80   | 0.81      | 4.50    | 0.95    |
| II     | 4.65   | 4.67      | 1.00    | 5.26   | 1.13    | 5.84   | 1.26      | 6.00    | 1.29    |
| III    | 5.10   | 5.10      | 1.00    | 5.30   | 1.04    | 6.10   | 1.20      | 5.90    | 1.16    |
|        | 4.82 ± 0.24 | 4.78 ± 0.28 | 1.00 ± 0.02 | 4.80 ± 0.28 | 1.00 ± 0.16 | 5.25 ± 0.24 | 1.09 ± 0.24 | 5.47 ± 0.84 | 1.14 ± 0.17 |
| 0.005  |        |           |         |        |         |        |           |         |        |
| I      | 4.94   | 5.20      | 1.05    | 4.88   | 0.99    | 5.30   | 1.07      | 5.40    | 1.09    |
| II     | 4.57   | 4.94      | 1.08    | 4.50   | 0.98    | 4.83   | 1.06      | 4.99    | 1.09    |
| III    | 4.74   | 5.05      | 1.07    | 4.69   | 0.99    | 5.07   | 1.07      | 5.21    | 1.10    |
|        | 4.75 ± 0.19 | 4.69 ± 0.19 | 1.07 ± 0.01 | 4.69 ± 0.19 | 0.99 ± 0.01 | 5.07 ± 0.24 | 1.07 ± 0.21 | 5.20 ± 0.21 | 1.09 ± 0.01 |
| 0.01   |        |           |         |        |         |        |           |         |        |
| I      | 5.49   | 6.02      | 1.10    | 7.43   | 1.35    | 7.53   | 1.37      | 6.60    | 1.20    |
| II     | 4.94   | 5.54      | 1.12    | 7.17   | 1.45    | 6.96   | 1.41      | 5.97    | 1.21    |
| III    | 4.97   | 5.55      | 1.12    | 6.57   | 1.32    | 6.76   | 1.36      | 6.40    | 1.29    |
| eq (T) | 0 min | 0 min + T | 15 min | 45 min | 100 min |
|-------|-------|-----------|--------|--------|---------|
|       | FL / a.u. | FL / a.u. | F/F_0  | FL / a.u. | F/F_0  | FL / a.u. | F/F_0  |
| avg   | 5.13 ± 0.31 | 5.70 ± 0.27 | 1.11 ± 0.01 | 7.06 ± 0.44 | 1.38 ± 0.07 | 7.08 ± 0.40 | 1.38 ± 0.03 | 6.23 ± 0.32 | 1.23 ± 0.05 |
| 0.02  | I      | 4.83 | 5.65 | 1.17 | 7.18 | 1.49 | 7.30 | 1.51 | 7.00 | 1.45 |
|       | II     | 5.29 | 5.88 | 1.11 | 7.30 | 1.38 | 7.30 | 1.38 | 6.70 | 1.27 |
|       | III    | 5.49 | 6.41 | 1.17 | 7.64 | 1.39 | 8.05 | 1.47 | 8.22 | 1.50 |
| avg   | 5.20 ± 0.33 | 5.98 ± 0.39 | 1.14 ± 0.03 | 7.37 ± 0.24 | 1.42 ± 0.06 | 7.55 ± 0.43 | 1.45 ± 0.07 | 7.31 ± 0.81 | 1.40 ± 0.12 |
| 0.1   | I      | 5.61 | 8.02 | 1.43 | 9.43 | 1.68 | 9.88 | 1.76 | 10.26 | 1.83 |
|       | II     | 5.34 | 7.90 | 1.48 | 9.55 | 1.79 | 9.66 | 1.81 | 9.70 | 1.82 |
|       | III    | 5.36 | 8.00 | 1.49 | 9.66 | 1.80 | 11.05 | 2.06 | 12.27 | 2.29 |
| avg   | 5.44 ± 0.15 | 7.97 ± 0.06 | 1.47 ± 0.03 | 9.55 ± 0.12 | 1.76 ± 0.07 | 10.20 ± 0.75 | 1.88 ± 0.16 | 10.74 ± 1.35 | 1.98 ± 0.27 |

References

[1] S. Kalhor-Monfared, C. Beauvireau, D. Scherman, C. Girard, Eur. J. Med. Chem. 2016, 122, 436–441.
[2] J. R. Kramer, T. J. Deming, Biomacromolecules 2012, 13, 1719–1723.
