Trace metals, organic carbon and nutrients in the Beidagang Wetland Nature Reserve, northern China

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Abstract

This study aimed to determine sediment contamination in the Beidagang Wetland Nature Reserve to describe atmospheric deposition of trace metals. We analyzed Hg, Cd, Pb, TOC, TN, TP, δ¹³C, and δ¹⁵N, and studied their variations in surface sediments and in the vertical profiles of sediment cores collected from the reserve. Evaluation of environmental trace metal contamination using sediment quality guidelines and geochemical background values indicated that the risk of metal pollution in the reserve sediments was relatively low. Concentrations of Hg, Cd, and Pb in the sediments were much lower than concentrations in sediment samples from Bohai Bay and polluted rivers in Tianjin. Enrichment factors indicate that samples are moderately contaminated with Hg, Cd, and Pb; whereas the geo-accumulation index results classify the sediments as uncontaminated to moderately contaminated with Hg, Cd, and Pb. The distribution patterns of trace metal concentrations in the three core samples were uniform.

δ¹³C and δ¹⁵N were used to track the sources of TOC and TN in sediments. Results show that TOC mainly originated from the residue and decaying matter of aquatic plants (e.g., algae, reeds, and Typha), while TN was derived from soil N and elevated atmospheric N deposition. Because domestic and industrial waste is not discharged into the Beidagang Wetland Nature Reserve, trace metals found in sediments mainly originate from atmospheric deposition. The results provide baseline data for analysis of trace metal accumulation in Beijing-Tianjin-Hebei, a region subject to atmospheric deposition in northern China.

Introduction

Beijing-Tianjin-Hebei is the largest urbanized region in northern China, comprising the economic region surrounding the cities of Beijing, Tianjin, and Hebei, along the coast of the Bohai Sea. In this region, increasing energy production, domestic heating, and automobile traffic, have resulted in the release of considerable volumes of trace metals into the atmosphere. Thus, air quality is a major environmental issue affecting sustainable urban development and the health of inhabitants in this region. Soils in China have been contaminated with heavy...
metals to varying degrees. Cd and Hg have been identified as priorities for control due to their higher concentrations in soils and higher public health risks [1]. Several trace metals (e.g., Hg, Cd, and Pb) are known to be atmospheric pollutants, highly toxic to humans and wildlife even at extremely low levels. Trace metals are transported in the atmosphere either as gases or by adsorption onto particulate matter [2] and are widely distributed in the natural environment. Trace metal emissions into the atmosphere, and subsequent dry/wet deposition either directly to soils or to the surface of water bodies are important potential sources of surface water contamination in Beijing-Tianjin-Hebei. Over the last 10 years, several studies have investigated the sources and impacts of trace metal contamination in the atmosphere [3–6] and soils [7,8] of this region. However, investigations of trace metal concentrations in wetland nature reserve sediments from northern China have been limited. Wetland nature reserves are important sources of data because they are protected from human activities once they are designated as reserves, and thus sewage and wastewater cannot be discharged within their boundaries. Sediment samples from wetland nature reserves are therefore important for assessing the levels and potential sources of trace metals that have resulted from atmospheric deposition in the Beijing-Tianjin-Hebei region.

The objectives of this study were to: (1) Evaluate trace metal pollution and accumulation in the Beidagang Wetland Nature Reserve sediments; (2) Identify potential sources, and spatial and temporal variations in total organic carbon (TOC), total nitrogen (TN), total phosphorus (TP), carbon and nitrogen stable isotopes (δ13C and δ15N), Hg, Cd, and Pb inputs to the reserve, and; (3) Assess the influence of atmospheric deposition of trace metals within the reserve.

Materials and methods

Ethics statement

No specific permissions were required for the present study as we only collected a limited number of water and sediment samples from the nature reserve. Our field study did not involve any endangered or protected species.

Study area

The Beidagang Wetland Nature Reserve (38°36′–38°57′ N and 117°11′–117°37′ E) is situated on the coast of Bohai Bay, southeast of Tianjin, northern China. The reserve has an area of about 34,887.13 ha and falls on a migration route vital for the annual movement of birds between the north and south of East Asia. It is an important wintering and staging area for more than 100,000 birds. The reserve is divided into three functional zones: the core region, buffer zone, and experimental zone. The core region of the reserve was founded in 1974 and has an area of about 16,400 ha [9]. Although considerable ecological changes have occurred in and around the reserve in recent years due to increased salinization and water resource shortages [10], assessment of anthropogenic impacts on the environment through quantification of trace metals (i.e., Hg, Cd, and Pb), organic carbon, and nutrients in the Beidagang Wetland Nature Reserve sediments remains very limited.

Samples

Surface sediment samples were collected in September 2014 from the top 0–2.5 cm at 12 sites and sediment profile samples were collected from the top 0–35 cm at three sampling sites, all located within the core zone of the Beidagang Wetland Nature Reserve (Fig 1). Sediment cores were then sliced at 2.5 cm intervals and immediately frozen. All samples were air-dried at 50°C overnight, passed through a 2mm stainless steel sieve to remove large debris, and then
homogenized by quartering with a riffler. The resulting subsamples were ground and sieved so that they could pass through a 100-mesh sieve for analysis.

**Chemical analyses**

For Hg, Cd, Pb, and Fe analyses, 0.20 g of sediment was weighed in a 55mL fluorocarbon (TFM) microwave vessel, 5 mL aqua regia was added, and less than 5 min was required for the
completion of the specific reactions. The tubes were then tightly sealed. The mixture was then digested for 10 min using microwave heating in a suitable laboratory microwave unit (MARS 5, CEM). In this process, the temperature of each sample should rise to 180°C in less than 5.5 min and remain between 175°C and 185°C for the remaining 10 min irradiation period. After cooling, the content was diluted to 50 mL with Milli-Q water. The Hg content was determined using cold vapor atomic fluorescence spectrometry while the solution was limpid. The Cd, Pb, and Fe concentrations were measured using inductively coupled plasmamass spectrometry with a SCIEX Elan 9000(Perkin-Elmer, USA) equipped with a Ryton cross-flow nebulizer.

The sediment samples were air-dried, ground and passed through a 2-mm sieve. The pH was determined from the <2 mm fraction in a slurry of 1:2.5 sediment: water suspension using a calibrated ORION 1260 Ion Selective Electrode (Thermo Fisher Scientific, USA). The salt content was determined from the <2 mm fraction in a slurry of 1:5 sediment: water suspension using a conductivity meter. An aliquot of sediment (5 g wet weight) was stirred for 1 h with HCl 6 M under constant N2 flux, and the S2− ions trapped in a 0.5 M NaOH solution were measured with methylene blue spectrophotometric method. Sodium acetate and conductance methods were used to determine the cation exchange capacity (CEC) and salt content of the samples. TN was determined using an elemental analyzer (Vario EL/MICRO cube, Elementar, Hanau, Germany). Sulfuric-perchloric acid digestion of sediments was used for TP determination, and TP was analyzed using a spectrophotometer (Tu-1810, Persee, China) through the Mo-Sb colorimetric method.

For determination of TOC and isotope composition, 5 g of sediment was weighed and soaked with 2 mol/L HCl to remove carbonate. The mixture was then washed using ultra-pure water to remove HCl, prior to freeze-drying and grinding. TOC was determined using an elemental analyzer (Vario EL/MICRO cube, Elementar, Hanau, Germany). δ15N and δ13C were determined using a Nu Horizon continuous-flow isotope ratio mass spectrometer (Nu Instruments, Wrexham, UK) and a perspective stable isotope ratio mass spectrometer (Nu Instruments). IAEA-CO-9 barium carbonate and IAEA-NO-3 potassium nitrate were used as standards for the calculations of δ13C and δ15N, respectively. Variations in 15N versus 14N, and 13C versus 12C were expressed as ‰ deviations relative to their reference standards in δ units, as follows:

\[
\delta^{15}N(\text{‰}) = \left( \frac{R_{\text{Sample}}}{R_{\text{Standard}}} - 1 \right) \times 1000, \tag{1}
\]

and;

\[
\delta^{13}C(\text{‰}) = \left( \frac{R_{\text{Sample}}}{R_{\text{Standard}}} - 1 \right) \times 1000. \tag{2}
\]

In these expressions, R is the corresponding ratio of either 15N/14N or 13C/12C.

**Quality control**

In order to check analytical quality, sample preparation, and instrument performance, laboratory quality control was conducted via analyses of a certified reference sediment (i.e., sediment GBW-07310: 0.28±0.03, 1.12±0.08, 26±2, 720±90,271±23, and 400±40 mg/kg for total Hg, Cd, Pb, N, P, and TOC, respectively). All control samples were prepared and analyzed identically to the study samples. In this study, GBW-07310 reference total Hg, Cd, Pb, N, P, and TOC concentrations were 0.29±0.03, 1.20±0.09, 25±2.1, 643±72, 287±22, and 348±56 mg/kg respectively in three replicates.
Statistical analysis
All data were analyzed using Microsoft Excel 2007 and SPSS version 12.0, and differences were considered statistically significant at \( p < 0.05 \). All data in this study are shown as the mean ± standard deviation \((n = 3)\). Relationships between variables were based on Pearson’s correlation coefficients, and factor analysis was applied to group pollutants using SPSS 12.0. The statistical significance of observed differences between metals, TOC, and nutrients released in each sediment core was determined using ANOVA in SPSS 12.0. Mapping for each sample site was performed using ArcGIS 10.0 (ESRI).

Metal accumulation assessment
Geochemical background concentrations of metals and NOAA (National Oceanic and Atmospheric Administration) sediment quality guidelines were applied to assess heavy metal accumulation and sediment pollution levels. The enrichment factor \((EF)\), based on the normalization of a tested element against a conservative reference element, is widely used to estimate anthropogenic impacts on sediments. The geo-accumulation index \((I_{geo})\) is another commonly applied index for assessing heavy metal contamination by comparing measured metal concentrations with local background levels [11]. In this study, the local background levels are the metal concentrations observed in preindustrial sediments. The \( EF \) and \( I_{geo} \) indices are defined as follows:

\[
EF = \frac{C_n / C_{Ref}}{B_n / B_{Ref}}_{\text{Sample}}
\]

\[
I_{geo} = \log_2 \left( \frac{C_n}{1.5B_n} \right)
\]

In these equations, \( C_n \) is the measured metal concentration in the samples, \( C_{Ref} \) is the reference element concentration in the study environment, \( B_n \) is the geochemical background concentration of the metal and \( B_{Ref} \) is the reference element concentration in the reference environment. The value 1.5 in Eq 2 is a background matrix correction factor that takes into account possible variations in values resulting from lithogenic effects. A detailed explanation of \( EF \) and \( I_{geo} \) is presented in Abuduwaili et al. (2015)[11]. In the present study, Fe was used as the reference element for geochemical normalization. Background soil geochemical concentrations in Tianjin were reported as 0.05 mg/kg, 0.09 mg/kg, 20.32 mg/kg, and 3.49% for Hg, Cd, Pb, and Fe respectively, by Wu and Cao (2010)[12].

Results and discussion
Sediment characteristics
Surface sediment samples from the Beidagang Wetland Nature Reserve were separated into four fractions: clay (<2 \( \mu m \), 17.1%), silt (2–20 \( \mu m \), 21.9%), fine sand (20–200 \( \mu m \), 45.3%), and coarse sand (>200 \( \mu m \), 15.7%). The physicochemical sediment characteristics are summarized in Table 1. The average pH of the surface sediment samples was 7.8, lower than measured for the bottom sediments (pH = 8.1). The average TOC, TN, and TP concentrations were 1.38%, 0.115%, and 0.049%, respectively in the surface sediment samples. All nutrient concentrations were higher than measured in the bottom sediment samples (0.66%, 0.056%, and 0.021%). In
addition, TOC/TN and $\delta^{13}$C values in the surface sediments were 10.66 and –21.6 %, respectively, which were lower than the corresponding values (12.13 and –20.5%) in the bottom sediments. The TOC/TN ratio of organic matter is indicative of the source of organic matter; TOC/TN ratios exceeded 10 in the sediment core samples, which indicates the input of terrigenous organic matter [13].

The capacity of sediments to store and accumulate trace metals depends mainly on their CEC and adsorption capacity, which vary with clay content, types of clay minerals, organic matter, and oxide or hydroxide content. Results show that CEC in the surface sediment samples (13.4 meq/100g) was less than that in the bottom sediments (20.1 meq/100g), while $\delta^{15}$N in surface sediments (0.93‰) was greater than that in bottom sediments (–0.02‰). The total salt content in the surface sediment samples (1.91%) was greater than the corresponding value for bottom sediments (0.88%). The bottom sediments had the highest CEC values and adsorption capacity, while the surface sediments had the highest total salt content, which means higher mobility of Cd is expected due to higher total salt content.

**Trace metal accumulation and distribution**

The concentrations of Hg, Cd, and Pb in the samples are shown in Table 1. Compared with the NOAA sediment quality guidelines [14], samples from the Beidagang Wetland Nature Reserve can be considered lowly polluted because they do not exceed the Effects Range-Low (ERL) values of 0.15 mg/kg, 1.2 mg/kg, and 46.7 mg/kg for Hg, Cd, and Pb, respectively. Metal concentrations in the sediment samples were also lower than the sediment quality guideline threshold (Threshold Effects Levels, TEL) and Probable Effects Levels (PEL) of 0.17 mg/kg and 0.486 mg/kg for Hg, 0.6 mg/kg and 3.53 mg/kg for Cd, and 35 mg/kg and 91.3 mg/kg for Pb, respectively. Measured values below these standard levels suggest that the health of wildlife is not at risk from trace metal contamination and show that sediment pollution is a minor concern in the Beidagang Wetland Nature Reserve.

Concentrations of Hg and Cd in surface sediments collected from the polluted rivers in Tianjin (the Beitang River, Haizhu River, Haihe River, and Dagu River) exceeded the ERL and TEL values and thus represent a health risk to wildlife. Domestic and industrial waste was directly discharged into these rivers, and high trace metal concentrations were derived from sewage and wastewater effluents [15–17]. These data show that the concentrations of Hg and Cd in the Beidagang Wetland Nature Reserve are relatively low.

### Table 1. Physicochemical characteristics of sediment cores.

|                     | Surface sediment(0–20 cm) | Bottom sediment(20–35 cm) |
|---------------------|--------------------------|---------------------------|
| pH                  | 7.8                      | 8.1                       |
| Water content(%)    | 34.3                     | 23.7                      |
| TOC(%)              | 1.38                     | 0.66                      |
| TN(%)               | 0.115                    | 0.056                     |
| TP(%)               | 0.049                    | 0.021                     |
| TOC/TN              | 10.66                    | 12.13                     |
| $\delta^{13}$C(‰)  | -21.6                    | -20.5                     |
| $\delta^{15}$N(‰)  | 0.93                     | -0.02                     |
| Cl (%)              | 0.11                     | 0.06                      |
| Total salt content (%) | 0.19                    | 0.09                      |
| Hg(mg/kg)           | 0.114                    | 0.088                     |
| Cd(mg/kg)           | 0.160                    | 0.177                     |
| Pb(mg/kg)           | 30.0                     | 23.8                      |

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The Beidagang Wetland Nature Reserve is becoming more saline by the combined effects of sea level rise and altered hydrological and climatic conditions [9,10]. The ions of Ca\(^{2+}\), Mg\(^{2+}\), and Cl\(^{-}\) were also measured in the surface sediments at concentrations as high as 0.35, 0.82, and 1.09 g/kg, respectively. Hg and Cd can easily combine with Cl\(^{-}\) ion to form HgCl\(_4^{2-}\) and CdCl\(_2\). The stability and solubility of these complexes were higher than the affinity of Hg and Cd for the solid sediment phase. Furthermore, substantial amounts of Hg, Cd, and Pb associated with the dissolved fraction could also be easily leached from the sediments under the attack of ions such as Ca\(^{2+}\) and Mg\(^{2+}\). These might increase the mobility of heavy metals and would result in the metals finally being lost through water induced leaching.

Enrichment factor and \(I_{\text{geo}}\) index results for sediment trace metals from the Beidagang Wetland Nature Reserve are presented in Fig 2. These data show that overall the sediment cores from different sites have similar \(EF\) and \(I_{\text{geo}}\) values for Hg, Cd, and Pb. The average \(EF\) values for Cd, Hg, and Pb were 2.98, 2.54, and 2.03, respectively, which indicates moderate contamination with these metals (2 \(\leq\) \(EF\) \(<\) 5). Average \(I_{\text{geo}}\) values for Cd, Hg, and Pb were 0.65, 0.42, and 0.09, respectively, which indicates that sediments in the reserve range from uncontaminated to moderately contaminated (0 \(<\) \(I_{\text{geo}}\) \(\leq\) 1). Tang et al (2013) found that Cd, Cr, Cu, and Pb concentrations in Beijing urban street dust were elevated in comparison with Beijing soil background values. In this study, \(I_{\text{geo}}\) levels ranging from 0 to 5 were found with about 80\% of samples below the moderately polluted level [18]. Trace metal pollution universally occurred in the soil, street dust, and sediment in Beijing-Tianjin-Hebei region [4,8,12].

![Fig 2. Sediment core trace metal EF and \(I_{\text{geo}}\) results from the Beidagang Wetland Nature Reserve. The box extends from the lower quartile (25%) to the upper quartile (75%), covering the median (line) and the mean (black point) values; Vertical bars represent the error.](https://doi.org/10.1371/journal.pone.0204812.g002)
Vertical profiles of Hg, Cd, and Pb are shown in Fig 3. The average concentration of Hg was 0.07 mg/kg at approximately 35 cm depth, close to the sediment geochemical background concentration (0.05 mg/kg) for Hg in Bohai Bay. The concentration profile of the three sediment cores shows an increase in Hg concentration from the base of the profile to a height of approximately 17.5 cm, followed by a fairly uniform distribution from a depth of approximately 15 cm to the surface. The Hg concentration at a depth of approximately 10 cm was 0.14 mg/kg. Sediment profile results also show that Cd concentrations decrease from the surface to a depth of approximately 12.5 cm, increase from 15 cm to about 25 cm, and remain fairly uniform in distribution from about 27.5 cm to the bottom. The average Cd concentration (0.177 mg/kg) in bottom sediments (0–20 cm) was greater than the corresponding value (0.160 mg/kg) for surface sediments (0–20 cm). Previous studies showed that an increase in salinity caused by any salt promoted a higher and faster release of Cd in sediments than other metals (Pb, Cu, and Zn). The main mechanism regulating Cd mobility was the formation of Cd-chlorides [19,20]. Concentrations of Cd were 0.17 mg/kg at approximately 35 cm depth, close to the sediment geochemical background value (0.14 mg/kg) for Cd in Bohai Bay. Finally, profiles show that Pb concentrations decrease from the top to approximately 12.5 cm depth, then remain fairly uniform in distribution across the profile. Lead concentrations were equal to 20.7 mg/kg at approximately 35 cm, again close to the sediment geochemical background value (16.6 mg/kg) for Pb in Bohai Bay. A previous study reported Pb contents in a Bohai Sea sediment cores (0–35 cm) ranging between 23.65 mg/kg and 28.33 mg/kg (mean value: 25.14 mg/kg) [21].

There was no statistically significant difference between the concentrations of Hg, Cd, and Pb measured in cores 1, 2, and 3 (ANOVA, P>0.05). There were obvious temporal variations but no significant spatial differences in trace metal concentrations of core sediments in the Beidagang Wetland Nature Reserve.
Sources of carbon and nitrogen

Sediment profiles of water content, TOC, TN, and TP are shown in Fig 4. All results show fairly uniform distributions (mean values of 46.6%, 2.97%, 0.19%, and 0.055%, respectively) from the surface to a depth of approximately 7.5 cm. Below 7.5 cm there is a sudden and abrupt change, with values decreasing by more than half (mean values of 25.2%, 0.74%, 0.08%, and 0.029%, respectively). There was no statistically significant difference between the water content, TOC, TN, or TP measured in cores 1, 2, and 3 (ANOVA, \( P > 0.05 \)).

Sediment profiles for TOC/TN, \( \delta^{13}C \), and \( \delta^{15}N \) are also shown in Fig 4. There is a significant difference in the \( \delta^{15}N \) values between the sediment cores (ANOVA, \( P < 0.05 \)), while the TOC/TN ratios and \( \delta^{13}C \) values show no significant difference. Elemental ratios such as TOC/TN are commonly used to indicate the biodegradability of dissolved organic matter [22,23]. These results indicate that TOC is similarly degraded in each core, and TOC and TN did not originate from sewage discharge.
Previous studies have shown that the TOC/TN ratio for fresh algal organic matter is between 8 and 3 and that the ratio for terrigenous organic matter from vascular higher plants is around 20 or higher [24]. Thus, a TOC/TN ratio in sediment organic matter (SOM) greater than 8 is thought to be indicative of influence from these two sources. If the TOC in terrigenous SOM accounts for a higher proportion, the TOC/TN ratio will be even higher. As a result, residues and decaying matter of aquatic plants (e.g., algae, reeds, and Typha) are considered the two main sources of SOM in the Beidagang Wetland Nature Reserve. The TOC/TN values in this study range from 6.66 to 22.60 (average: 12.00 ± 3.62) and are significantly higher than those calculated in a previous study (range: 6.73 to 9.78; average: 7.73 ± 0.750) of the Yongdingxin River estuary (ANOVA, p = 0.001) [25]. Thus, higher TOC/TN values may reflect lower anthropogenic inputs in the Beidagang Wetland Nature Reserve compared to the Yongdingxin River estuary.

As shown in Figs 3 and 4, the profiles of TOC and TN are significantly different to the profiles of Hg, Cd, and Pb. Variations in Hg and Cd concentration in sediment cores were also different, and results show that the trace metals (Hg, Cd, and Pb), TOC, and nutrients do not exhibit either common behaviors or sources in the sediments. The Hg and Cd concentration profiles in the three sediment cores show a slow increase from the bottom to the top, implying that the concentrations at the base of cores may be natural values (i.e., geochemical background concentrations).

Fig 5 shows the TOC concentration compared to TN, and δ¹⁵N compared to δ¹³C. These data show that TOC and TN are significantly correlated (p < 0.01), indicating that they share common behaviors or sources (Fig 5A). Bulk sedimentary stable isotopes of organic carbon (δ¹³C) and TOC/TN ratios have been widely and successfully used as markers to estimate changes in the relative proportions of terrigenous and marine organic matter in coastal sediments [26].

Fig 5B shows that δ¹³C values for sediments in the Beidagang Wetland Nature Reserve ranged from –24.8‰ to –18.7‰ (average: –21.2±2.38‰). Box plots of δ¹³C and δ¹⁵N values from various sources are shown in Fig 6. Marine organic matter is typically enriched in ¹³C compared to terrestrial plant material because of differences in photosynthetic mechanisms and carbon sources. Terrestrial C3 vascular plants and their C4 counterparts employ different photosynthetic pathways and thus produce different δ¹³C values. A δ¹³C value in the range of C3...
(-28% to -26%) and C4 plants (-14% to -12%) as well as a C/N ratio greater than 12 indicates a terrestrial end member, while a δ¹³C in the range -22% to -19% and a C/N ratio less than 8 suggests an aquatic end member[27]. Therefore, results indicate that TOC in the study sediments is not predominantly derived from terrestrial and soil organic matter. Potential alternative sources include residues and decaying matter of aquatic plants (e.g., algae, reed, and Typha).

Results show that δ¹⁵N values for sediments in the Beidagang Wetland Nature Reserve range from -0.911% to 3.11% (Fig 5B). Increases in reactive nitrogen are attributed to intensive land use, the increased use of N-containing organic and inorganic fertilizers, animal manure, discharge of human sewage, and elevated atmospheric N deposition[31–33]. Organic and inorganic fertilizers (i.e., ammonium, NO₃⁻, and urea) have typical δ¹⁵N values between -6% and +6%, while typical δ¹⁵N values for atmospheric N deposition are between -13% and +13%. Variation within this range is controlled by complex chemical reactions in the atmosphere as well as a variety of anthropogenic sources including fossil fuel combustion. Manure and sewage are enriched in δ¹⁵N relative to other N sources; δ¹⁵N values that originate from manure are between +5% and +25%, while those from sewage are between +4% and +19%. Typical δ¹⁵N values for soil nitrogen range between 0% and +8% (Fig 6). Results show that TN in the Beidagang Wetland Nature Reserve sediments mainly originates from soil nitrogen and elevated atmospheric nitrogen deposition. Thus, fertilizers, sewage, and manure were not the main sources of nitrogen in this system, indicating that sewage and wastewater do not discharge into the nature reserve.

Results show that δ¹³C values in the Beidagang Wetland Nature Reserve sediments range from -24.8% to -18.7% (average: -21.2±2.38%), significantly higher than the range -23.7% to -22.5% previously reported for the Yongdingxin River estuary sediments (average: -23.1±0.35%) (ANOVA, p = 0.001)[25]. If δ¹³C values are in the range -22% to -16%, organic matter could be from a mixture of terrigenous and aquatic components. One possible explanation for this result is that phytoplankton prefers to absorb CO₂ in ¹³C enriched air which results in a decreased δ¹³C value. In contrast, the δ¹⁵N values in this study ranged from -0.911% to 3.11% (average: 1.16±1.24%), significantly less than the range (-2.53% to 6.77%,

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**Fig 6.** Box plots of δ¹³C and δ¹⁵N values from various sources. The lower and upper edges of the box represent 25th and 75th percentiles, encompassing the mean value (line); the lower and upper bars represent 10th and 90th percentiles. (Data from the references [28–30]; data in Beidagang SOM from the present study).
average: 2.28±0.11% reported for Yongdingxin River estuary sediments (ANOVA, \(p = 0.004\)) and suggesting a slightly different enrichment of N in these samples. The low \(\delta^{15}N\) reported in the nature reserve sediments might be due to contributions from soil nitrogen, as terrestrial plant ecosystems typically have lower \(\delta^{15}N\). Lower \(\delta^{15}N\) values may also reflect lower anthropogenic input of nitrogen into the Beidagang Wetland Nature Reserve compared with the Yongdingxin River estuary. These results show that the sedimentary environment of the Beidagang Wetland Nature Reserve has not been seriously impacted by human activities such as the discharge of wastewater and sewage. Thus, the accumulation of trace metals has likely been the result of atmospheric deposition.

**Multivariate statistical analysis**

Major sources of trace metal pollution in aquatic ecosystems are domestic wastewater effluents, coal-burning power plants, nonferrous metal smelters, iron and steel plants, and dumping of sewage sludge. The atmosphere is the major route for Pb entering natural waters [34]. Principal component analysis (PCA) was performed to further identify statistically significant sources influencing trace metal contamination in the Beidagang Wetland Nature Reserve. For this analysis, only PCs with eigenvalues greater than 1 were considered important; the analysis revealed that three factors were responsible for 71.55% of the total variation. The PCA results, with variable loadings, variance, and PCs, are presented in Table 2 and Fig 7.

The results show that component 1 (PC1) has a total variance of 35.83% and is substantially positively loaded with TOC, TN, TP, and \(\delta^{13}C\). All four of these variables are generated from the residue and decaying organic matter of aquatic plants. The \(\delta^{13}C\) value in plants is close to the values resulting from atmospheric deposition and soils. In contrast, PC2 is responsible for 22.84% of the total variation, and a moderate amount of Hg and Cd is loaded into the sediments. This is significant because atmospheric deposition is an important source of Hg and Cd[8].

The trace metals Hg and Cd are significant pollutants that influence soil quality in Beijing and Tianjin. Previous research has demonstrated that coal burning and weathering of HgS, used as a pigment in ancient buildings, significantly contributes to higher Hg concentrations [35]. Coal combustion has been identified as the major source of particulate mercury in aerosol samples in Beijing [6]. Because there is a potential link between atmospheric particles and contaminated soils, Schleicher et al. (2016) compared atmospheric particulate Hg concentrations in total suspended particulates samples from Beijing with soil guideline values for the

**Table 2. Loadings of measured variables in the Beidagang Wetland Nature Reserve, northern China.**

| Parameters | PC1      | PC2      | PC3      |
|------------|----------|----------|----------|
| TOC(%)     | 0.81     | 0.47     | 0.14     |
| TN(%)      | 0.88     | 0.37     | 0.05     |
| TP(%)      | 0.80     | 0.02     | 0.19     |
| TOC/TN     | -0.21    | 0.67     | 0.38     |
| \(\delta^{13}C(\%)\) | -0.51    | 0.41     | 0.47     |
| \(\delta^{15}N(\%)\) | 0.50     | 0.03     | -0.64    |
| Hg(mg/kg)  | 0.33     | 0.65     | -0.12    |
| Cd(mg/kg)  | -0.54    | 0.59     | -0.26    |
| Pb(mg/kg)  | 0.46     | -0.57    | 0.50     |
| Eigenvalue | 3.22     | 2.06     | 1.16     |
| %Total variance | 35.83 | 22.84  | 12.88    |
| Cumulative total variance | 35.83 | 58.67  | 71.55    |

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Their results showed that average particulate Hg concentrations in August were 1.03±0.21 mg/kg between 2006 and 2010, 2.24±0.85 mg/kg in December between 2005 and 2010, and 2.11±0.66 mg/kg in January between 2006 and 2011. These values are of the same magnitude as the soil guideline value of 1.00 mg/kg Hg for residential areas in the UK, and the Chinese Grade II guideline.

Results show that PC3 has a total variance of 12.88% and is responsible for the positive loading of Pb into the Beidagang Wetland Nature Reserve sediments. Lead mainly originates from traffic contaminants, and therefore Pb is likely to have come from anthropogenic sources including vehicular traffic and industrial discharge, as reported in previous work on soils in Tianjin [8]. In another related study, Shi et al. (2008) identified the source of metals in urban dust using geostatistical and multivariate analyses, reporting that Pb mainly originates from traffic contamination and Cd largely comes from point-source industrial pollution [37]. Levels of Pb and Cd in urban dust were higher than in suburban dust, while concentrations of Hg were higher in suburban dust, indicating a different main source [38].

Pearson’s correlation analysis has often been employed to determine the common source (and/or carrier substances) of trace metals. Thus, in the present study, a correlation matrix was calculated for the elements in question to discuss sediment responses to anthropogenic contaminants in the Beidagang Wetland Nature Reserve (Table 3). Results show that TOC is positively correlated with TN, TP, and Hg, while TN is significantly correlated with TP, δ¹⁵N, and Hg. Fe is positively correlated with Pb and TP. Results also indicate that TOC, TN, and TP may have common sources. A positive linear correlation was observed between TOC and Hg (r = 0.38, P = 0.021, n = 51) (Fig 8), while Cd was not significantly correlated with TOC. Results also indicate that Hg and Cd have different behaviors and sources. Divalent metals such as Cd and Pb have a high affinity with sulfide in anoxic sediment, while sulfide is not correlated with Hg, Cd, and Pb in the Beidagang Wetland Nature Reserve. The sediments were aerobic due to the lack of water resources and the low water level in the wetland. Sulfide might become oxidized to sulfate. A significant negative correlation (r = -0.63, p < 0.01) between Cd and Pb also shows that these metals have different behaviors. However, a positive correlation...
between these trace metals in contaminated sediments of western Bohai Bay and in adjacent estuaries have been reported in previous studies [39,40]. Correlations were significant in rivers and estuaries affected by human activities while the correlations were poor in the Beidagang Wetland Nature Reserve.

Previous studies have shown that Hg in sediments in polluted rivers and estuaries in Tianjin are significantly correlated with TOC, and not positively correlated with Cd [16,40]. The total Hg concentrations in the Haihe River and Dagu Drainage River sediments were significantly correlated with TOC, TN, TP, δ¹³C, δ¹⁵N, Fe, Hg, Cd, and Pb.

![Graph](https://doi.org/10.1371/journal.pone.0204812.g008)

**Fig 8.** Hg concentrations compared to TOC concentrations.

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|       | TOC  | TN   | TP    | Sulfide | TOC/TN | δ¹³C  | δ¹⁵N  | Fe      | Hg      | Cd      | Pb      |
|-------|------|------|-------|---------|---------|-------|-------|---------|---------|---------|---------|
| TOC   | 1    | 0.97a| 0.58a | 0.32    | 0.24    | -0.22 | 0.28  | -0.29   | 0.38b   | -0.16   | 0.13    |
| TN    | 1    | 0.62a| 0.15  | 0.02    | -0.32   | 0.34b | 0.16  | 0.38b   | -0.23   | 0.19    |
| TP    | 1    | 0.41 | -0.15 | -0.32   | 0.21    | 0.37b | 0.30  | -0.38b  | 0.38b   |
| Sulfide | 1   | 0.25 | -0.29 | 0.26    | -0.20   | 0.02  | 0.19  | 0.22    |
| TOC/TN| 1    |      | 0.39b | -0.18   | -0.32   | 0.12  | 0.31  | -0.31   |
| δ¹³C  | 1    |      | -0.34b| 0.09    | 0.21    | 0.29  | -0.17 |
| δ¹⁵N  | 1    |      | 0.27  | 0.28    | -0.23   | 0.01  |
| Fe    | 1    |      | 0.09  | 0.26    | 0.43b   |
| Hg    | 1    |      | 0.18  | -0.13   |
| Cd    |      |      | 1     | -0.63a  |
| Pb    |      |      | 1     |         |

*a* Correlation is significant at the 0.01 level (2-tailed).

*b* Correlation is significant at the 0.05 level (2-tailed).

Table 3. Correlation coefficients for TOC, TN, TP, TOC/TN, δ¹³N, δ¹⁵C, Hg, Cd, and Pb.

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correlated with TOC contents ($r = 0.87$, $p = 0.99$, $n = 21$) [16]. This result is also in agreement with a previous study on Haihe River sediments; a strong correlation between Hg and TOC concentrations was found in the heavily polluted river sediments, indicating that concentrations of Hg were significantly influenced by sediment TOC contents [15]. A recent investigation indicated that Hg was significantly correlated with TOC and the correlation explained 78% of the variance in the data; this investigation showed that Hg was associated with organic matter in water bodies [41]. All these results suggest that sediment organic matter plays an important role in the sedimentary accumulation of Hg in wetland ecosystems. Although organic matter plays the most important role in the distribution of trace metals in sediments, its role is not well understood [23]. For example, Sanei and Goodarzi reported that the strong affinity between Hg and soluble organic matter is due not only to its chemical reactivity but also to the physical characteristics of these labile compounds [42].

**Atmospheric wet or dry deposition**

The Beijing-Tianjin-Hebei region has a very large consumption of energy and power, the majority of which is sourced from coal combustion, identified as the major source of particulate Hg in Beijing and Tianjin [5,6]. Total global Hg emissions from anthropogenic sources to the atmosphere range from 1900 to 2900 t/yr; East Asia currently accounts for almost 40% of total global anthropogenic emissions of Hg [43]. Total anthropogenic Hg emissions in China were estimated to have continuously increased from 356 t in 2000 to 538 t in 2010 with an average annual increase rate of 4.2% [44]. Coal-fired power plants in China emit less than 100 tons Hg directly to the atmosphere every year due to nationwide air control actions [45]. The Beijing, Tianjin, and Hebei provinces combusted 23, 53, and 302 million tons of coal in 2013, respectively, and total coal combustion was estimated to be 378 million tons/yr in this region. Coal combustion from coal-fired power plants was 13.37, 24.2, and 89.58 million tons/yr in these provinces, respectively in 2013 [46]. The emission factors for Hg, Cd and Pb vary in the ranges of 20–430 mg/ton coal, 1.25–13.11 mg/ton coal, and 307.39–2965.73 mg/ton coal, respectively [47,48]. Atmospheric trace metals from coal combustion are deposited on the ground and on vegetation before being brought into water circulation. Atmospheric emissions represent a major pathway for trace metals to enter the surface environment. Wet and dry deposition of atmospheric Hg is important sources of Hg to terrestrial ecosystems in the nature reserve.

In recent decades, there has been increasing research interest in air pollutants in the Beijing-Tianjin-Hebei region. However, only a few studies have considered the atmospheric deposition of trace metals. The particulate Hg concentration in aerosol samples from Beijing between January and December 2006 was 573±551 ng/m³ [4]. The Hg concentrations in snow samples from Tianjin in three heating seasons in winter (2012–2015) ranged from 1.72 to 3.03 μg/L, and the Cd concentrations ranged from 0.15 to 0.72 μg/L [49]. Wet deposition of Hg in Beijing was 101.52 μg/(m² yr) (Nov 1994–Nov 1995) and 123.09 μg/(m² yr) (Nov 1995–Nov 1996) [50]. The atmospheric dry deposition fluxes of Hg and Cd in Tianjin were 35.3–37.5 μg/(m² yr) and 97.4–103.6 μg/(m² yr), respectively [5]. Atmospheric deposition of Hg in the Bohai Sea was 31.0 μg/(m² yr) [51]. The atmospheric wet depositions of Pb in Mount Tai and Jiaozhou Bay, northern China were 7570 and 2210 μg/(m² yr), respectively, while the depositions of Cd were 340 and 130 μg/(m² yr), respectively [52]. A previous study reported that the annual flux of atmospheric Hg deposition in China’s adjacent seas was 140 tons. The proportion of Hg that was deposited into sediment was 50% of the total Hg output. Hg evasion from surface sediment and surface water comprised 2.4% of the total Hg input and 26% of the total Hg output of the seas [51]. Industrial and urban areas have high anthropogenic emissions of
Hg and are regarded as Hg output areas. In contrast, terrestrial ecosystems in nature reserves, which have much lower Hg emissions and are exposed to high Hg loadings from polluted areas, are sinks of atmospheric Hg and experience Hg accumulation.

The concentrations of Hg and Cd in topsoil from Tianjin in 2013 were 0.40 μg/g and 0.18 μg/g, respectively [8], while Hg and Cd concentrations in topsoil from Beijing in 2015 were 0.40 μg/g and 0.72 μg/g, respectively [35,53]. The concentrations were of the same magnitude as the values in the Tianjin and Beijing topsoil. These concentrations were much higher than the Tianjin soil background values, 0.05 μg/g for Hg and 0.09 μg/g for Cd. Atmospheric deposition was identified as an important source of trace metals, especially Hg and Cd, in soils and the ecosystem in the Beijing-Tianjin-Hebei region.

Conclusions

Combined analysis of trace metals, and stable carbon and nitrogen isotopic composition of organic matter was used to assess the origin and accumulation in sediments from the Beidagang Wetland Nature Reserve. Based on the enrichment factor and the geo-accumulation index calculated in the present study, the sediments can be considered to be unpolluted to moderately polluted with Hg, Cd, and Pb. However, compared to other metal-polluted rivers in Tianjin, concentrations of Hg and Cd in the sediments were relatively low. There were obvious temporal variations but no significant spatial differences in trace metal concentrations in the sediment cores. Stable isotopes of carbon and nitrogen indicate that urban sewage and industrial waste do not discharge into the Beidagang Wetland Nature Reserve, and thus the potential sources of TOC in this area are residue and decaying aquatic plant matter (e.g., algae, reed, and Typha). TN mainly originates from the soil and elevated atmospheric nitrogen deposition. Atmospheric deposition is the main source of trace metals in sediments from the Beidagang Wetland Nature Reserve. Trace metal concentrations should be monitored to provide basic information about the impact of atmospheric deposition in Tianjin. Our results provide a baseline dataset for trace metal accumulation predating the industrialization of Tianjin to the present day for key locations. This can be used for comparisons with other regions to assess the effects of atmospheric deposition on trace metal contamination in northern China.

Supporting information

S1 Data. Complete primary data file.

Author Contributions

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References

1. Zhang X, Zhong T, Liu L, Ouyang X. Impact of soil heavy metal pollution on food safety in China. PLoS ONE. 2015; 10(8): e0135182. https://doi.org/10.1371/journal.pone.0135182 PMID: 26252956
2. Pavageau MP, Pechevan C, Krupp EM, Morin A, Donard OFX. Volatile metal species in coal combustion flue gas. Environ Sci Technol. 2002; 36: 1561–1573. https://doi.org/10.1021/es015595s PMID: 11999087
3. Okuda T, Katsuno M, Naoi D, Nakao S, Tanaka S, He K, et al. Trends in hazardous trace metal concentrations in aerosols collected in Beijing, China from 2001 to 2006. Chemosphere. 2008; 72: 917–924. https://doi.org/10.1016/j.chemosphere.2008.03.033 PMID: 18462775

4. Schleicher NJ, Norra S, Chai F, Chen Y, Wang S, Cen K, et al. Temporal variability of trace metal mobility of urban particulate matter from Beijing–A contribution to health impact assessment of aerosols. Atmos Environ. 2011; 45: 7248–7265. https://doi.org/10.1016/j.atmosenv.2011.08.067

5. Wei Z, Wu GH, Su RX, Li CW, Liang PY. Mobility and contamination assessment of mercury in coal fly ash, atmospheric deposition, and soil collected from Tianjin, China. Environ Toxicol Chem. 2011; 30: 1997–2003. https://doi.org/10.1002/etc.605 PMID: 21713969

6. Schleicher NJ, Schafer J, Blanc G, Chen Y, Chai F, Cen K, et al. Atmospheric particulate mercury in the megacity Beijing: Spatiotemporal variations and source apportionment. Atmos Environ. 2015; 109: 251–261. https://doi.org/10.1016/j.atmosenv.2015.03.018

7. Chen X, Xia X, Wu S, Wang F, Guo X. Mercury in urban soils with various types of land use in Beijing, China. Environ Pollut. 2010; 158: 48–54. https://doi.org/10.1016/j.envpol.2009.08.028 PMID: 19765869

8. Zhao L, Xu YF, Hou H, Shangguan YX, Li FS. Source identification and health risk assessment of metals in urban soils around the Tanggu chemical industrial district, Tianjin, China. Sci Total Environ. 2014; 468: 654–662. https://doi.org/10.1016/j.scitotenv.2013.08.094 PMID: 24061056

9. You P, Li H-H, Wang S-X. The diversity of the moth community in the North Dagang Wetland Nature Reserve, Tianjin. Acta Ecologica Sinica. 2006; 26(4): 999–1004. (In Chinese)

10. Wei Q, Sun CH, Wu GH, Pan L. Haihe River discharge to Bohai Bay, North China: trends, climate, and human activities. Hydro Res. 2017; 48(4): 1058–1070. https://doi.org/10.2166/nh.2016.142

11. Abuduawili J, Zhang ZY, Jiang FQ. Assessment of the distribution, sources and potential ecological risk of heavy metals in the dry surface sediment of Aibi Lake in Northwest China. PLoS ONE. 2015; 10(3): e0120001. https://doi.org/10.1371/journal.pone.0120001 PMID: 25781032

12. Wu GH, Cao SS. Mercury and cadmium contamination of irrigation water, sediment, soil and shallow groundwater in a wastewater-irrigated field in Tianjin, China. Bull Environ Contam Toxic. 2010; 84(3): 336–341. https://doi.org/10.1007/s00128-010-9939-6 PMID: 2111947

13. Hyun S, Lee T, Jung S-K. Contents and stable isotopic compositions of organic carbon and total nitrogen in the surface sediment of two coastal bays in Korea. Ocean Sci J. 2011, 46: 289–297. https://doi.org/10.1007/s12601-011-0022-y

14. Long E, Macdonald D, Smith S, Calder F. Incidence of adverse biological effects within ranges of chemical concentrations in marine and estuarine sediments. Environ Manag. 1995; 19: 81–97. https://doi.org/10.1007/BF02472006

15. Wang XL, Sato T, Xing BS, Tao S. Health risks of heavy metals to the general public in Tianjin, China via consumption of vegetables and fish. Sci Total Environ. 2005; 350: 28–37. https://doi.org/10.1016/j.scitotenv.2004.09.044 PMID: 16227070

16. Shi JB, Liang LN, Jiang GB, Jin XL. The speciation and bioavailability of mercury in sediments of Haihe River, China. Environ Int. 2005; 31: 357–365. https://doi.org/10.1016/j.envint.2004.08.008 PMID: 15734189

17. Qiao J, Shao DZ, Luo SM, Duan L, Hua GY, Zhang CD. Pollution characteristics of heavy metals in sediment from Haizhu River in Tianjin Binhai New Development Area and selection of regional heavy metal pollution indicators. Res Environ Sci. 2010; 23(1): 1343–1350 (In Chinese)

18. Tang RL, Ma KM, Zhang YX, Mao QZ. The spatial characteristics and pollution levels of metals in urban street dust of Beijing, China. Appl Geochem. 2013; 35: 88–98. https://doi.org/10.1016/j.apgeochem.2013.03.016

19. Laing GD, Vos RD, Vandecasteele B, Lesage E, Tack FMG, Verloo MG. Effect of salinity on heavy metal mobility and availability in intertidal sediments of the Scheldt estuary. Estuar Coast Shelf Sci. 2008; 77(4): 589–602. https://doi.org/10.1016/j.ecss.2007.10.017

20. Acosta JA, Jansen B, Kalbitz K, Faz A, Martínez-Martínez S. Salinity increases mobility of heavy metals in soils. Chemosphere, 2011; 85(8): 1318–1324. https://doi.org/10.1016/j.chemosphere.2011.07.046 PMID: 21862104

21. Liu M, Fan DJ, Zheng SW, Tian Y, Zhang AB. Tracking lead origins in the central Bohai Sea based on stable lead isotope composition. Haiyangxueba o. 2016; 2: 36–47 (In Chinese)

22. Fellman JB, D’Amore DV, Hood E, Boone RD. Fluorescence characteristics and biodegradability of dissolved organic matter in forest and wetland soils from coastal temperate watersheds in southeast Alaska. Biogeochemistry. 2008; 88: 169–184. https://doi.org/10.1007/s10533-008-9203-x

23. Elbishlawi H, Jaffe PR. Characterization of dissolved organic matter from a restored urban marsh and its role in the mobilization of trace metals. Chemosphere. 2015; 127: 144–151. https://doi.org/10.1016/j.chemosphere.2014.12.080 PMID: 25681788
24. Emerson S, Hedges JL. Processes controlling the organic carbon content of open ocean sediments. Paleoceanography. 1988; 3(5): 621–634. https://doi.org/10.1029/PA003i005p00621

25. Sun CH, Wei Q, Ma LX, Pan L, Li L, Wu GH. Trace metal pollution and carbon and nitrogen isotope tracing through the Yonqingxian River estuary in Bohai Bay, Northern China. Mar Pollut Bull. 2016; 115: 451–458. https://doi.org/10.1016/j.marpolbul.2016.10.066 PMID: 27823829

26. Meyers PA. Preservation of elemental and isotopic source identification of sedimentary organic matter. Chem Geol. 1994; 144: 289–302. https://doi.org/10.1016/0009-2541(94)90059-0

27. Middelburg JJ, Nieuwenhuize J. Carbon and nitrogen stable isotopes in suspended matter and sediments from the Schelde Estuary. Mar Chem. 1998; 60: 217–225. https://doi.org/10.1016/S0304-4203(97)00104-7

28. Smith BN, Epstein S. Two categories of $^{13}$C/$^{12}$C ratios for higher plants. Plant Physiol. 1971; 47: 380–384. https://doi.org/10.1104/pp.47.3.380 PMID: 16657626

29. Sun CH, Wei Q, Guo C-C, Wang Z-L. Foliar stable carbon isotope ratios of Phragmites australis and the relevant environmental factors in marsh wetlands in Tianjin China. J Plant Ecol. 2015; 39(11): 1044–1052. https://doi.org/10.17521/cjpe.2015.0101

30. Ledgard SF, Wook C, Bergersen FJ. Isotopic fractionation during reduction of nitrate and nitrite by extracts of spinach leaves. Aust J Plant Physiol. 1985; 12: 631–640. https://doi.org/10.1071/pp9850631

31. Högb erg P, Högb erg MN, Quist M, Ekblad A, Näsholm T. Nitrogen isotope fractionation during nitrogen uptake by ectomycorrhizal and non–mycorrhizal Pinus sylvestris. New Phytol. 1999; 142: 569–576. https://doi.org/10.1046/j.1469-8137.1999.00404.x

32. Pacey JA, Trevor Scholtz M, Li Y-F. Global budget of trace metal sources. Environ Rev. 1995; 3: 145–159. https://doi.org/10.1080/71360-1385(01)01889-1 PMID: 11239611

33. Schleicher NJ, Schaefer J, Blanc G, Chen Y, Bi CJ, et al. Atmospheric particulate mercury in the megacity Beijing: Efficiency of mitigation measures and assessment of health effects. Atmos Environ. 2016; 124: 396–403. https://doi.org/10.1016/j.atmosenv.2015.08.039

34. Shi GT, Chen ZL, Xu SY, Zhang J, Wang L, Bi CJ, et al. Potentially toxic metal contamination of urban soils and roadside dust in Shanghai, China. Environ Pollut. 2008; 156: 251–260. https://doi.org/10.1016/j.envpol.2008.02.027 PMID: 18703261

35. Shi GT, Chen ZL, Bi CJ, Wang L, Teng JY, Li YS, et al. Potentially toxic metal contamination of urban soils and roadside dust in Beijing, China, since 1987. J Geochem Explor. 2013; 124: 195–202. https://doi.org/10.1016/j.jgecxpl.2012.09.007

36. Evans RD. Physiological mechanisms influencing plant nitrogen isotope composition. Trends Plant Sci. 2001; 6: 121–126. https://doi.org/10.1016/S1360-1385(01)01889-1 PMID: 11239611

37. Cheng H, Zhao C, Liu F, Yang K, Liu Y, Li M, et al. Mercury drop trend in urban soils in Beijing, China, since 1987. J Geochem Explor. 2013; 124: 396–403. https://doi.org/10.1016/j.jgecxpl.2012.09.007

38. Hoggberg P, Hoggberg MN, Quist M, Ekblad A, Nasholm T. Nitrogen isotope fractionation during nitrogen uptake by ectomycorrhizal and non–mycorrhizal Pinus sylvestris. New Phytol. 1999; 142: 569–576. https://doi.org/10.1046/j.1469-8137.1999.00404.x

39. Evans RD. Physiological mechanisms influencing plant nitrogen isotope composition. Trends Plant Sci. 2001; 6: 121–126. https://doi.org/10.1016/S1360-1385(01)01889-1 PMID: 11239611

40. Wu GH, Shang JM, Pan L, Wang ZL. Heavy metals in surface sediments from nine estuaries along the coast of Bohai Bay, Northern China. Mar Pollut Bull. 2014; 82: 194–200. https://doi.org/10.1016/j.marpolbul.2014.02.033 PMID: 24650542

41. Guentze JL. Wetland influences on mercury transport and bioaccumulation in South Carolina. Sci Total Environ. 2009; 407: 1344–1353. https://doi.org/10.1016/j.scitotenv.2009.03.036 PMID: 19036414

42. Schleicher NJ, Schaefer J, Blanc G, Chen Y, Chai F, Cen K, et al. Atmospheric particulate mercury in the megacity Beijing: Efficiency of mitigation measures and assessment of health effects. Atmos Environ. 2016; 124: 396–403. https://doi.org/10.1016/j.atmosenv.2015.08.039

43. Si R, Chen ZL, Xu SY, Zhang J, Wang L, Bi CJ, et al. Potentially toxic metal contamination of urban soils and roadside dust in Shanghai, China. Environ Pollut. 2008; 156: 251–260. https://doi.org/10.1016/j.envpol.2008.02.027 PMID: 18703261

44. Wang L, Wang Z, Wu Y, Duan L, Wu Q, et al. Updated emission inventories for speciated atmospheric mercury from anthropogenic sources in China. Environ Sci Technol. 2015; 49: 3185–3194. https://doi.org/10.1021/es504840m PMID: 25655166

45. Tang SL, Wang LN, Feng XB, Feng ZH, Li RY, Fan HP, et al. Actual mercury speciation and mercury discharges from coal-fired power plants in Inner Mongolia, Northern China. Fuel. 2016; 180: 194–204. https://doi.org/10.1016/j.fuel.2016.04.037
46. Tian HZ, Wang Y, Xue ZG, Qu YP, Chai FH, Hao JM. Atmospheric emissions estimation of Hg, As, and Se from coal fired power plants in China, 2007. Sci Total Environ. 2011; 409: 3078–3081. https://doi.org/10.1016/j.scitotenv.2011.04.039 PMID: 21621816

47. Streets DG, Hao JM, Wu Y, Jiang JK, Chan M, Tian HZ, et al. Anthropogenic mercury emissions in China. Atmos Environ. 2005; 39: 7789–7806. https://doi.org/10.1016/j.atmosenv.2005.08.029

48. Deng S, Shi YJ, Liu Y, Zhang C, Wang XF, Cao Q, et al. Emission characteristics of Cd, Pb and Mn from coal combustion: Field study at coal-fired power plants in China. Fuel Process Technol. 2014; 126: 469–475. https://doi.org/10.1016/j.fuproc.2014.06.009

49. Wu GH, Wei Q, Sun CH, Gao JJ, Pan L, Guo L. Determination of major and trace elements in snow in Tianjin, China: a three-heating-season survey and assessment. Air Qual Atmos Health. 2016; 9: 687–696. https://doi.org/10.1007/s11869-015-0375-y

50. Liu J-H, Wang W-H, Peng A. Regional pattern of mercury wet deposition in Beijing. J Environ Sci. 2000; 12(S):77–84.

51. Liu MD, Chen L, Wang XJ, Zhang W, Tong YD, Ou LB, et al. Mercury export from Mainland China to adjacent seas and its influence on the marine mercury balance. Environ Sci Technol. 2016; 50: 6224–6232. https://doi.org/10.1021/acs.est.5b04999 PMID: 27243109

52. Xing JW, Song JM, Yuan HM, Wang OD, Li XG, Li N, et al. Atmospheric wet deposition of dissolved trace elements to Jiaozhou Bay, North China: Fluxes, sources and potential effects on aquatic environments. Chemosphere. 2017; 174: 428–436. https://doi.org/10.1016/j.chemosphere.2017.02.004 PMID: 28187389

53. Wei X, Gao B, Wang P, Zhou HD, Lu J. Pollution characteristics and health risk assessment of heavy metals in street dusts from different functional areas in Beijing, China. Ecotox Environ Safety. 2015; 112: 186–192. https://doi.org/10.1016/j.ecoenv.2014.11.005 PMID: 25463870