C/SiOC Composites by a Modified PIP Using Solid Polysiloxane: Fabrication, Microstructure and Mechanical Properties

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Abstract: A modified polymer infiltration and pyrolysis method (PIP) was developed to enhance the densification of C/SiOC composites, using molten MK resin (polymethylsilsesquioxane) as precursor. Organic sulfonic acid was added as cross-linking agent to lower the curing temperature. The cross-linking mechanism and ceramization behavior of MK resin was studied. A high ceramic yield of about 85wt% and low free carbon content below 3wt% are achieved, with excellent high-temperature stability of the derived SiOC. The modified PIP approach presents a high densification efficiency. After only 8 PIP cycles, the final C/SiOC composites possess a density of about 1.81 g/cm³. Compared with those composites fabricated by conventional PIP process, the C/SiOC composites prepared by modified PIP process show a much denser microstructure with much more improved densification efficiency. Bending strength of the as-fabricated C/SiOC composites is of (312±25) MPa, showing obvious non-brittle fracture behavior.

Key words: C/SiOC composite; high-efficient PIP; microstructure; cross-linking mechanism

Continuous fiber reinforced ceramic matrix composites (CFRCMCs) have been considered as excellent structure materials for the applications requiring low specific weight, high strength and toughness at elevated temperature[1-2]. Generally, several approaches can be used for the fabrication of CFRCMCs, such as chemical vapor infiltration (CVI), slurry infiltration (SI), and precursor infiltration and pyrolysis (PIP), etc.[3-4]. Among these methods, the PIP route is getting increasing attention in recent years, due to its advantages in low processing temperature, controllable ceramic compositions, suitable for large-scale components fabrication with complicated shapes and low cost[5-6].

As a type of amorphous ceramic, SiOC can be considered as an anionic modification of silica glass in which Si–O bonds are partially replaced by Si–C bonds, and the other carbon atoms, free carbon, form graphene-like layers[7], which are expected to improve the thermal and mechanical properties for the applications at high temperatures[8-9]. Owing to the advantages such as excellent high temperature stability, oxidization resistance and low cost, polymer-derived silicon oxycarbides (SiOC) ceramics attract lots of attention being of CRFCMCs. As reported, C/SiOC composites showed excellent mechanical performance, which was fabricated by using polysiloxane (PSO) as SiOC precursor[10]. In recent years, liquid hydrogen-containing PSO or polymethyl(phenyl)siloxane flake was used as precursor for SiOC. Their pyrolysates at 1000 °C have relatively high content of free carbon (about 26wt% or even 29wt%)[11-12], which can easily react with SiO₂ phase at high temperature, leading to high-temperature instability of composites. Among the common polysiloxanes, MK resin is a carbon-poor precursor for SiOC ceramic with carbon content of about 14wt% and only about 8wt% free carbon[13-14], which makes MK resin a promising precursor for C/SiOC composites.

In the present study, a modified PIP process with high efficiency was developed to fabricate C/SiOC composites based on MK resin. The cross-linking and pyrolysis behavior of MK resin was optimized and the cross-linking mechanism was studied. The microstructures and mechanical properties of C/SiOC composites fabricated through the modified PIP process were investigated.
1 Experimental

A commercially available polysiloxane (MK resin, Wacker GmbH, Burghausen, Germany) was used as the precursor for SiOC matrix. As reported in literature\cite{15-17}, MK resin in state of white powders, possesses ca. 2mol% of cross-linkable hydroxy and ethoxy groups with $[\text{CH}_3\text{SiO}_{1.5}]_n$ basic structure referred as polymethylsilsesquioxane. Its possible structure is depicted in Fig. 1\cite{15}.

To lower the curing temperature and shorten the curing period, organic sulfonic acid was chosen as crosslinking agent (henceforth referred as CLA).

MK resin was firstly compacted into a cylindroid tablet with a diameter of 10 mm, to investigate its softening behavior. The photograph and the size of the tablet were recorded \textit{in situ} as a function of temperature by a high temperature real-time observation and test system (TOM-AC, FRAUNHOFER-GESSELLSCHAFT, Germany) at a heating rate of 1 ℃/min in vacuum. The viscosity of MK resin at different temperatures and the dynamic modulus at 160 ℃ were measured by a rheometer (MCR301, Anton Paar GmbH, Austria). In order to investigate the curing mechanism, the functional groups of PSO before and after cross-linking process were analyzed by a FT-IR spectroscopy (NICOLET Is10, Thermo Scientific, America). To investigate ceramization behavior of MK resin, the thermogravimetric (TG) and differential thermal analysis (DTA) were performed with MK resin on a STA 449/F3 simultaneous thermal analyzer (Netzsch, Germany). The heating rate for DTA measurement was 10 ℃/min in flowing Ar. Content of oxygen and carbon atoms were analyzed by oxygen/ nitrogen determinator (TC600C, LECO, America) and carbon/ sulfur determinator (CS2000, ELTRA, Germany), respectively.

Three-dimensional needle-punched carbon fiber preforms fabricated with T700SC carbon fiber (12 K) were used as the reinforcement in this study. MK resin with different contents of CLA was cross-linked at 160 ℃ in vacuum and then pyrolyzed at 1000 ℃ in Ar atmosphere. A modified PIP process illustrated in Fig. 2 was performed to fabricate the C/SiOC composites. Following this process, the composites with dense matrix can be achieved. In the modified PIP process, MK resin with CLA was firstly heated to melt, and then was infiltrated into the fiber preform placed in a flexible container with aids of pressure difference between inside and outside of the flexible container. Secondly, the infiltrated preform was cured at 160 ℃ for 6 h. Thirdly, the cured specimens were pyrolyzed at 1000 ℃ for 30 min at a heating rate of 5 ℃/min in Ar atmosphere, and then cooled down to room temperature. PIP process was repeated for 8 times to get C/SiOC composites with high densities. As a contrast, a conventional PIP process with ethanol solution of MK resin being used as precursor was utilized to fabricate C/SiOC composites. To clearly distinguish between the two kinds of C/SiOC composites, the C/SiOC composites fabricated by the modified PIP process is henceforth referred as M-C/SiOC composites, and the C/SiOC composites fabricated by the conventional PIP process is referred as C-C/SiOC composites.

The densities as well as open porosities of the C/SiC composites were measured by Archimedes method with deionized water as the immersion medium. The bending strength of the final composites was characterized by three-point bending test (DDL20, Changchun Research Institute for Mechanical Science Co. Ltd, Changchun, China) with a span length of 50 mm and a crosshead speed of 0.5 mm/min. The dimensions of the specimens for three-point bending test were 60 mm×6 mm × 4 mm. Microstructures of composites were observed on both the fracture surfaces and polished cross-sections using a field-emission scanning electron microscope (Hitachi SU8220, Japan).

![Fig. 1 Scheme of the possible structure of MK resin\cite{15}](image1)

![Fig. 2 Schematic diagram of the modified PIP process](image2)
2 Results and discussion

2.1 Viscosity-temperature characteristics of MK resin

The viscosity of precursor is a key parameter for control of the PIP process. The high viscosity of precursor would lead to a poor fluidity during the impregnation process, while the excessively low viscosity would make the precursor easily flow out of the preforms during the crosslinking process\[18\]. Therefore, the viscosity of MK resin was investigated firstly as a function of temperature. Softening point was determined by the change of geometric size of the MK resin as shown in Fig. 3(a). It can be observed that there is no obvious deformation of MK resin at temperature below 55 °C, while the right angle turned into round ones when the temperature was rising above 60 °C. It reveals that the softening point of MK resin is in the range of 55-60 °C. As shown in Fig. 3(b), it is clear that the viscosity of MK resin drops sharply at temperature higher than 110 °C. The viscosities of MK resin are about 100 mPa·s at 150 °C, 144 mPa·s at 160 °C and 87 mPa·s at 170 °C, respectively.

2.2 Curing and ceramization behavior of MK resin

As a kind of thermoplastic polymer, MK resin begins to cross-link above 250 °C, while the low viscosity can lead to waste of precursor during the pyrolysis. To lower the curing temperature, CLA was applied as crosslinking agent considering the cross-linkable hydroxy and ethoxy groups.

CLA of various contents (1wt%, 0.5wt%, 0.1wt%, 0.05wt%, and 0.01wt%) were added into MK resin to pick out the most suitable ratio. The optical photographs of MK resin with different contents of CLA heated at 160 °C are shown in Fig. 4, and the weight losses are also labeled. It can be observed that when no CLA was added, MK resin kept in the liquid state even after being heated at 160 °C for 12 h and the weight loss is only 0.67wt%. With the addition of CLA, MK resin can convert into solid state and the weight loss increased, which meant CLA could effectively promote the cross-linking of MK resin. When the content of CLA was higher than 0.1wt%, MK resin can get cross-linked within 30 min. When 1wt% CLA was added, the cross-linking time was as short as 5 min and the weight loss increased to 3.32wt%. When the content of CLA decreased to 0.05wt% and 0.01wt%, the cross-linking time increased to about 1 and 3 h, respectively. It can also be observed that when more than 0.05wt% CLA was added, the cross-linked precursor presented a porous structure, which may be caused by escaping of generated gases during the cross-linking.

Rheological parameters such as storage modulus ($G'$) and loss modulus ($G''$) were employed to further evaluate the curing behavior of MK resin. For the sake of convenience, MK resin with 0.05wt% CLA was selected for rheological test. As shown in Fig. 5, storage moduli and loss moduli kept constant at the beginning of the test and then increased slowly until 1000 s. Two moduli increased sharply after 1000 s, and there was an intersection point of $G'$ and $G''$ at about 1970 s. After the intersection point, the $G''$ exceeded $G'$, which indicated that the elastic deformation of MK resin was dominant rather than viscous deformation of materials. Furthermore, the polymer began to transform into elastomer. The gel point of MK resin, on which the polymer begins to crosslink, can be determined by the intersection point of $G'$ and $G''$[19]. It can be concluded that MK resin with 0.05wt% CLA began to crosslink after being treated at 160 °C for 0.5 h, which is in consistent with the phenomenon referred above.

To investigate the curing mechanism, FT-IR spectroscopy of the as-received MK resin (without cross-linking) and MK resin with 0.01wt% CLA cured at 160 °C were tested (Fig. 6). The spectrum of MK resin shows absorption bands at $\nu=765, 1275$ cm$^{-1}$ (Si–CH3), $\nu=2913, 2975$ cm$^{-1}$ (C–H), $\nu=1618, 3437$ cm$^{-1}$ (Si–OH), $\nu=1023$ cm$^{-1}$ (Si–O–C) and $\nu=1111$ cm$^{-1}$ (Si–O–Si), as previously
reported\cite{16,20-23}. As the curing time increases, the intensity of absorption bands assigned to Si–O–C and Si–O–Si slightly increases. What’s more, the intensities of Si–O–Si and Si–O–C increases with curing time, indicating that the network of Si–O–Si was gradually formed. It can be concluded that condensation reactions of Si–OH with Si–OH, Si–OEt as well as with RSO\textsubscript{2}–OH groups with the aids of CLA are main reactions involved during cross-linking.

FT-IR spectroscopy of the cured MK resins with different contents of CLA were also tested as shown in Fig. 7. There is no obvious difference in their IR spectra except for the absorption bands for Si–O (at 1023 and 1111 cm\textsuperscript{-1}) and –OH (at 1618 and 3437 cm\textsuperscript{-1}). As the CLA content increases, the intensity ratio of Si–O–Si and Si–O–C absorption increases, which reveals that CLA affects curing rate and cross-linking degree of MK resin. Excessively rapid cross-linking rate results in generation of...

Fig. 4  Optical photographs of MK resin with different contents of CLA after heat treatment at 160 ℃

Fig. 5  Storage moduli and loss moduli versus curing time for MK resin with 0.05wt% CLA at 160 ℃

Fig. 6  (a) Cross-linking reactions during the curing process; (b) FT-IR spectra of as-obtained MK resin without curing and MK resin with 0.01wt% CLA after curing at 160 ℃ for different time; (c) details in enlarged view of wavenumber in the range from 1250 cm\textsuperscript{-1} to 500 cm\textsuperscript{-1}
of a mass of small molecules such as ethanol and H₂O which fails to volatilize before Si–O–Si network is formed, and is then absorbed in porous structure of the cured MK resin. Consequently, the intensity of –OH bands show a remarkable increase.

TG-DTA curves of MK resin without CLA from 100 ℃ to 1000 ℃ were shown in Fig. 8. The ceramic yield is as high as 88wt% after pyrolyzation at 1000 ℃. The weight loss mainly occurred in two temperature ranges. A weight loss of about 4.91wt% occurred at temperatures ranging from 170 ℃ to 350 ℃, which may be ascribed to the vaporization of H₂O, n-propanol, ethanol and POSS during the cross-linking process [16,23-26]. A further weight loss of about ~6.09wt% occurs between 350 and 900 ℃, which may be resulted from breakage/rearrangement of molecular chains and release of a large number of small molecular gas, such as H₂, H₂O and propane, etc. [15,22]. Table 1 shows the ceramic yields of MK resin with different CLA contents pyrolized at 1000 ℃ in Ar atmosphere and the chemical compositions of the pyrolysates in the formation of xSiO₂·ySiC·zC. The ceramic yields of MK resin with different contents of CLA ranges from 82.9wt% to 85.6wt%, which means that the effect of CLA on the ceramic yield is negligible. It is indicated that free carbon is less than 3wt% in pyrolysates.

2.3 Microstructures and mechanical properties of C/SiOC composites

Based on the above-mentioned cross-linking and pyrolysis behavior of MK resin with different CLA contents, it is found that the MK resin with 0.01wt% of CLA is a suitable matrix precursor for the fabrication of C/SiOC composites, which can guarantee enough time for the molten precursor to be infiltrated into the fiber preform. During the PIP process, the precursor was first cross-linked at 160 ℃ for 4 h before pyrolysis.

Fig. 9 shows the density and the open porosity of C/SiOC composites as a function of PIP cycles. It is obvious that the density increases with the increase of PIP cycles, and the open porosity decreases. As a result of high ceramic yield of precursor as well as eliminating effect of solvent, the high efficiency in modified PIP was realized in comparation with the conventional ones. Remarkably, the open porosity of M-C/SiOC composites (6.2%) is much lower than that of C-C/SiOC composites (17.4%) after 8 cycles. After the first cycle, the density of M-C/SiOC composites increased from 0.6 g/cm³ to 1.14 g/cm³. Compared with the C-C/SiOC composites, the final M-C/SiOC composite possessed relatively high density of about 1.81 g/cm³ after the eighth cycle. Similarly, the density of C/SiOC composites fabricated via the modified PIP process in this work is increased by about 10% in comparation with the results of 1.64 g/cm³ in the reported literature [27].

Fig. 10 shows the cross-sectional microstructures of the final C/SiOC composites fabricated by two different PIP processes. Compared with the C-C/SiOC composites

| Table 1  The ceramic yield and the element composition for MK resin with different CLA contents pyrolized at 1000 ℃ |
|-----------------|---------|-------|-----|-------|-----------------|-------|
| CLA content/wt% | Ceramic yield/wt% | Si/wt% | O/wt% | C/wt% | Empirical formula | Free C/wt% |
|-----------------|---------|-------|-----|-------|-----------------|-------|
| 0               | 83.7    | 53.17 | 33.96 | 12.87 | 0.56SiO₂·0.44SiC·0.12C │ 2.82  |
| 0.01            | 85.1    | 53.13 | 33.89 | 12.98 | 0.56SiO₂·0.44SiC·0.13C │ 2.92  |
| 0.05            | 85.7    | 53.89 | 33.05 | 13.06 | 0.54SiO₂·0.46SiC·0.10C │ 2.36  |
| 0.1             | 84.7    | 54.46 | 33.15 | 13.39 | 0.54SiO₂·0.46SiC·0.13C │ 2.91  |
| 0.5             | 84.1    | 54.74 | 31.54 | 13.72 | 0.50SiO₂·0.50SiC·0.09C │ 2.09  |
| 1               | 82.9    | 53.86 | 32.57 | 13.57 | 0.53SiO₂·0.47SiC·0.12C │ 2.70  |
Fig. 9  Density and open porosity variation of C/SiOC composites during the PIP process

Fig. 10  SEM images of the polished cross sections of (a) C-C/SiOC composites; (b-d) M-C/SiOC composites

(Fig. 10(a)), the inter-bundle pores in M-C/SiOC composites is obviously much less (Fig. 10(b)). Although some pores can still be found in the matrices, the final M-C/SiOC composites are highly dense in general. When the preforms were infiltrated via the modified PIP process, the precursors with moderate viscosity can fill in almost all pores and turn into solid state during the cross-link process. The existence of several micro-pores and microcracks in matrices (Fig.10(b-d)) can be attributed to generation of close pores resulted from porous pyrolysates during the precice PIP cycles. Therefore, the precursor can’t reach anymore in the latter infiltration process.

The three-point-bending test shows that the M-C/SiOC composites possess a relatively higher bending strength of (312±25) MPa than the C-C/SiOC composites with a bending strength of (211±7) MPa. The M-C/SiOC composites also show superiority in mechanical properties as compared to the C/SiOC composites prepared by conventional PIP process with other polysiloxane precursors reported (217[28], 247[29] and 265[11]). The M-C/SiOC composites show a typical non-brittle fracture behavior, which can be concluded from the bending stress-strain curves of three-point bending test as shown in Fig. 11. The stress-strain curve of the M-C/SiOC composites can be divided into three regions, namely, linear region, non-linear region and fracture region. In the non-linear region, micro-cracks generation and propagation as well as the interfacial debonding are responsible for the pseudo-plastic deformation behavior[30]. Once the stress reaching to the maximum as observed in the fracture region, the stress drops abruptly. Then the stress show a step-like decrease, which may be ascribed to the layer-fiber-distribution characteristics of the preforms. At the beginning of three-point-bending test, fibers on the underside layer have the largest deformation. When fibers at this layer was destroyed, the second layer of fibers momentarily bear the load, therefore the imposed load dropped abruptly, then slightly increased again. Meanwhile, the C-C/SiOC composites show a narrower non-linear region, leading to the worse bending strength.

Fig. 12 shows the microstructure of fracture surfaces of the M-C/SiOC composites. Pulled-out fibers can be observed in the SiOC matrix. As the deposited PyC interphase ensures the weak bonding between matrix and fiber, cracks can propagate either along the matrix-PyC interface or along PyC-fiber interphase. Therefore, the de-bonding phenomenon can be observed on the surface of the pulled-out fibers.

3 Conclusion

Based on a modified highly-efficient PIP process using molten MK resin, C/SiOC composites with a density of 1.81 g/cm³ was fabricated by only 8 cycles of PIP.
When MK resin with 0.01 wt% CLA was firstly cross-linked at 160 °C for 4 h, precursor can transform into solid state and fill in both inter-bundle pores and intra-bundle pores, therefore only very small amount of micro-pores can be found in the composites. The as-fabricated C/SiOC composites possess a relatively high bending strength of (312±25) MPa and show an obvious non-brittle fracture behavior.

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基于固体聚硅氧烷的前驱体浸渍裂解法(PIP)制备 C/SiOC 复合材料及其微结构与力学性能研究

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摘 要: 以熔融的 MK 树脂(聚甲基倍半硅氧烷)为前驱体, 采用改进的前驱体浸渍裂解法(PIP)制备了致密的 C/SiOC 复合材料。为了降低 MK 树脂的固化温度选择有机磺酸作为交联剂，并采用红外光谱分析仪(FT-IR)和热重分析-差热分析仪(TG-DTA)研究了 MK 树脂的固化机理和陶瓷化行为。研究表明: MK 树脂的陶瓷产率高达 85wt%, 其裂解得到的 SiOC 陶瓷自由碳含量低于 3wt%, 有利于提高陶瓷的高温稳定性。经过 8 次 PIP 制备的 C/SiOC 复合材料的密度可达 1.82 g/cm³。对得到的 C/SiOC 复合材料进行三点弯曲测试, 其弯曲强度为(312±25) MPa, 表现出明显的非脆性断裂行为。

关 键 词: C/SiOC 复合材料; 高效 PIP; 微观结构; 交联机理

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