Logarithmic Partition Coefficient Comparison Study and Molecular Weight of Synthesized Prodrugs of Ibuprofen+Paracetamol, Diclofenac Sodium+Paracetamol and Ibuprofen+Diclofenac Sodium

Dhrubo Jyoti Sen* and Jalpa G. Patel

Department of Pharmaceutical Quality Assurance and Pharmaceutical Chemistry, Shri Sarvajanik Pharmacy College, Gujarat Technological University, Arvind Baug, Mehsana-384001, Gujarat, India

Date of Receipt- 05/11/2016
Date of Revision- 07/11/2016
Date of Acceptance- 15/11/2016

ABSTRACT

Prodrug is a substance which after administration is metabolized into a pharmacologically active drug. Actually prodrug has least medicinal value in in-vitro/in-vivo but after biotransformation by metabolism in in-vivo it releases the active medicament. A drug is a substance which is a chemical entity, has definite structural skeleton, obtained by natural or synthetic or semisynthetic source, which can fit on bioreceptor platform having controlling capacity to control over the biochemical malfunction. Every drug is xenobiotic because it is coming from outer source (xeno) and active in biological unit (biotic). Prodrug is the precursor of drug which is made by derivatization of the same to enhance the bioavailability by pharmacokinetics, lipid solubility by partition coefficient and increase the physicochemical and biochemical parameters by pharmacodynamics. Ibuprofen, Diclofenac and Paracetamol have been taken as NSAID (Non-Steroidal Anti Inflammatory Drug) and three prodrugs have been synthesized by reacting with acid chloride of ibuprofen and diclofenac with paracetamol to get prodrug of ester linkage and acid chloride of ibuprofen has been reacted with diclofenac to get prodrug of amide linkage. All three prodrugs showed different logP values and molecular weights according to the solubility parameters and electronegativity: logP profile: Prodrug-C>Prodrug-B>prodrug-A; molecular weight profile: Prodrug-C>Prodrug-B>prodrug-A.

Keywords: Ibuprofen, Diclofenac sodium, Paracetamol, Thionyl chloride, logP, pKa, Melting Point, Prodrug.

INTRODUCTION

Three prodrugs (Prodrug-A, Prodrug-B and Prodrug-C) have been successfully synthesized and have shown different melting points from individual parent drugs (Ibuprofen, Diclofenac sodium and Paracetamol) which indicates the authenticity of fulfillment of prodrug synthesis 1-2. The synthesized prodrugs were characterized by MP, logP values and IR (Infrared) spectrum for structural authentication. Their solubility parameters also found different from parent drugs. Prodrug is a substance having no medicinal importance but after biotransformation in GIT it releases the parent drug which is able to show the pharmacological activity. 3-5

Ibuprofen [C13H18O2; MW=206] (2-[4-(2-methylpropyl)phenyl]propanoic acid), from isobutylphenylpropanoic acid, is a drug in the nonsteroidal anti-inflammatory drug (NSAID) class used for treating pain, fever and inflammation. logP=3.97, pKa=4.91. Ibuprofen m.p.=theoretical (76°C), practical (75°C).
Diclofenac sodium \([NaC_14H_{10}Cl_2NO; MW=302]\) (Sodium \{2-[(2,6-dichlorophenyl)amino]phenyl\}acetic acetate) is a nonsteroidal anti-inflammatory drug (NSAID) taken or applied to reduce inflammation and as an analgesic reducing pain in certain conditions. \(\log P=4.51, pK_a=4.15\). Diclofenac sodium m.p.=theoretical (283-285°C), practical (282°C) Paracetamol \([C_8H_9NO_2; MW=151]\) (N-(4-hydroxyphenyl)acetamide), also known as acetaminophen or APAP, is a medication used to treat pain and fever. \(\log P=0.46, pK_a=9.38\). Paracetamol m.p.=theoretical (169°C), practical (170°C) (Figure 1).

Ibuprofen, Diclofenac and Paracetamol all comes under NSAID and all three are acidic in nature. Ibuprofen and diclofenac both have free carboxylic acid (–COOH) group and paracetamol has free phenolic group (–OH). The idea of formation of prodrug by joining of free carboxylic acid (–COOH) group of ibuprofen and diclofenac with phenolic group (–OH) of paracetamol by converting free carboxylic acid (–COOH) group into acid chloride (–COCl) and conjugated with free phenolic group (–OH) by benzoylation reaction to get the three desired compounds [Prodrug-A and Prodrug-B having ester (–COO–) linkage. Acid chloride (–COCl) of ibuprofen when reacts with imino group (–NH) of diclofenac then it produces Prodrug-C having amide (–CONH–) linkage (Figure 2).]

**Chemistry**

Ibuprofen (2 g, 0.01 m) has been reacted with 5 ml thionyl chloride and refluxed for 30 mins in moisture free environment until all ibuprofen has been dissolved. The reaction mixture was heated on water bath to remove excess thionyl chloride. It was then cooled in ice and paracetamol (1.5 g, 0.01 m) dissolved in methanol was added in it with shaking. Oily droplets separates out which was then mixed with ice water and methanol and kept in ice to solidify the droplets into white solid of prodrug-B. It was filtered and %yield and m.p. has been recorded for dried mass. Prodrug-B (\(\log P=4.90, \text{nonpolar}\)) is a combination of diclofenac sodium (\(\log P=4.51, \text{nonpolar}\)) and paracetamol (\(\log P=0.46, \text{polar}\)). MP=162-165°C, %yield=88.65.

Ibuprofen (2 g, 0.01 m) has been reacted with 5 ml thionyl chloride and refluxed for 30 mins in moisture free environment until all ibuprofen has been dissolved. The reaction mixture was heated on water bath to remove excess thionyl chloride. Diclofenac sodium (3 g, 0.01 m) was dissolved in acetone and added slowly to this acid chloride mixture with stirring in ice bath to get Prodrug-C after addition of water. Prodrug-C (\(\log P=6.13, \text{nonpolar}\)) is a combination of ibuprofen (\(\log P=3.97, \text{nonpolar}\)) and diclofenac sodium (\(\log P=4.51, \text{nonpolar}\)). MP=162-165°C, %yield=91.32. The beauty of reaction shows that diclofenac sodium which is sodium salt that releases free diclofenac during reaction between thionyl chloride in acid environment to get free –COOH group which then reacts with thionyl chloride to produce acid chloride [–COCl] which then combines with phenolic group [–OH] of paracetamol to give Prodrug-B [4–(acetylamino)phenyl \{2-[[2,6-dichlorophenyl]amino]phenyl\} acetate]. [Molecular Formula=C_{22}H_{18}Cl_{2}N_{2}O_{3}, Formula Weight=429.29]. (Solubility=0.1 g soluble in 10 ml methanol/10ml acetone) (Figure 3).

Ibuprofen, diclofenac sodium and paracetamol all are white in colour. Prodrug-A [Ibuprofen and Paracetamol] produces off white product. Prodrug-B [Diclofenac and Paracetamol] produces dark brown product. Prodrug-C [Ibuprofen and Diclofenac] produces light brown product after keeping in refrigerator overnight after addition of ethanol. Prodrug-C [N-(2,6-dichloro phenyl)-2-[(2S)-2-[4-(2-methylpropyl)phenyl]propanoyl]amino]phenyl]
acetic acid], MP=162-165°C, %yield=91.32. (Solubility=0.1 g soluble in 10 ml methanol/10 ml acetone). [Molecular Formula=C$_{27}$H$_{27}$Cl$_2$NO$_3$, Formula Weight=484.41]. This is due to formation of amide linkage (–CONH–) because ester linkage (–COO–) produced the products prodrug-A and prodrug-B within an hour. The linkage (–CO–X–; X=O/NH) shows the electronegativity (O=3.44 and N=3.04); so ester (–COO–) forms faster than amide (–CONH–) because O=3.44 > N=3.04.

**CONCLUSION**

Prodrug-A (logP=4.56) is formed from ibuprofen (logP=3.97)+paracetamol (logP=0.46): [4-(acetylamino)phenyl (2S)-2-[4-(2-methylpropyl) phenyl]propanoate].

[Molecular Formula=C$_{21}$H$_{25}$ NO$_3$, Formula Weight=339.42].

Prodrug-B (logP=4.90) is formed from diclofenac sodium (logP=4.51)+paracetamol (logP=0.46): [4-(acetylamino)phenyl 2-{[(2,6-dichlorophenyl)amino][phenyl]acetate].

[Molecular Formular=C$_{22}$H$_{18}$Cl$_2$N$_2$O$_3$, Formula Weight=429.29].

Prodrug-C (logP=6.13) is formed from ibuprofen (logP=3.97)+diclofenac sodium (logP=4.51): [N–(2,6-dichlorophenyl)[amino][phenyl]acetic acid].

[Molecular Formula=C$_{27}$H$_{27}$Cl$_2$NO$_3$, Formula Weight=484.41].

**logP profile:**

Prodrug-C>Prodrug-B>prodrug-A

**Molecular weight profile:**

Prodrug-C>Prodrug-B>prodrug-A

**Future Target**

Our future goal is to perform in-vitro biotransformation of these prodrugs by acidic and alkaline hydrolysis of both ester (–COO–) and amide (–CONH–) linkages into free drugs and chromatographically separation of their Rt in HPLC. Study of in-vitro biotransformation of prodrugs of ester and amide linkages of ibuprofen, diclofenac sodium and paracetamol in acidic and alkaline medium will be our next target after this project. Since both ester (–COO–) and amide (–CONH–) linkages are susceptible for hydrolysis in both acidic pH (gastric pH) and basic pH (intestinal pH) to produce parent drug ibuprofen, diclofenac and paracetamol by biotransformation; so it will be implemented as a prodrug which can show prolong action on pain and fever after getting release into free parent drug by biotransformation (Figure 4).

The HPLC (High Performance Liquid Chromatography) study will report the retention time (Rt) and release kinetics of three prodrugs by taking HPLC degradation datas of three samples of prodrugs and individual HPLC datas of parent drugs separately to compare the Rt, value of release of three drugs from prodrugs.

Prodrug-A (logP=4.56) will release Ibuprofen and Paracetamol, Prodrug-B (logP=4.90) will release Diclofenac sodium and Paracetamol and Prodrug-C (logP=6.13) will release Ibuprofen and Diclofenac. This will be a comparison study of drug release.

**REFERENCES**

1. Patel J G and Sen D J. Synthesis of prodrug of ester and amide linkages of NSAID having carboxylic acid, phenolic and imino groups: World Journal of Pharmacy and Pharmaceutical Sciences 2016; 5(11):897-908.

2. Vogel A I. A textbook of practical organic chemistry, 3rd edition, Longmans, Green and Co, Ltd, London, 1956; 791:361.

3. Slemmer J E, Martin B R and Damaj M I. Bupropion is a nicotinic antagonist. J Pharmacol Exp Ther 2000; 295(1):321-7.

4. Wu C, Quan J, Xie J, et al. Preparation and controlled release of degradable polymeric ketoprofen-saccharide conjugates. Polym Bull 2011; 67:593-608.

5. Babazadeh M and Mosanejhad T. Vinyl ester type polymers containing ibuprofen pendants: synthesis, characterization and evaluation. Iranian Polymer Journal 2009; 18(2):179-86.

6. Shargel L, Pong S W and Andrew B C. Applied Biopharmaceutics & Pharmacokinetics, 6th edition, McGraw-Hill Medical Publishing Division, US 2012; 47-66, 129-54.

7. Shirke S, Shewale S and Satpute M. Prodrug design: An overview. International Journal of Pharmaceutical Chemical and Biological Sciences 2015; 5(1):232-41.

8. Bhosale A V, Agrawal G P and Mishra P. Preparation and Evaluation of Directly Compressible Forms of Mutual Prodrugs of Ibuprofen Formuls of Mutual Prodrugs of Ibuprofen. Indian J Pharm Sci 2006; 68(4):425-31.
Figure 1. Prodrugs

Figure 2. Histogram of LogP of Prodrugs

Figure 3. Histogram of molecular weights of Prodrugs
Figure 4. Hydrolysis of Prodrugs