A comparative study on the evolution of the interface chemistry and electrical performance of ALD-driven Hf$_x$Ti$_y$Al$_z$O nanolaminates

Juan Gao, Gang He,* Lin Hao, Die Wang and Lin Zhao

In this work, a series of ternary HfTiO and TiAlO films and quaternary HfTiAlO films prepared with different stoichiometric ratios via atomic layer deposition (ALD) were deposited on Si substrates. The interfacial properties, band alignments, and electrical characteristics were analyzed to evaluate the electrical performance of the corresponding metal–oxide–semiconductor (MOS) capacitors. Based on the characterization data, it can be noted that the permittivity of HfTiO increases on increasing the Ti content, while the deterioration of the film quality leads to an increased leakage current. For TiAlO, as the Al content increased gradually, the interface quality improved, but the corresponding permittivity was sacrificed. As for HfTiAlO, the addition of Ti and Al jointly in HfO$_2$ could improve the interface performance, enlarge the energy band offset, enhance the dielectric constant, and reduce the leakage current simultaneously. The interfacial analysis and electrical characterization demonstrated that HfTiAlO with an ALD cycle ratio of Hf : Ti : Al = 6 : 1 : 1 had the most excellent film quality, interface performance and electrical properties, including a larger dielectric constant of 28.8, a larger conduction band offset with Si of 2.47 eV, and the lowest leakage current density of $1.11 \times 10^{-5}$ A cm$^{-2}$.

1. Introduction

With the development of semiconductor integrated circuits, the integration and performance of chips has been continuously improving. However, with the reduction in the characteristic size of the devices, the gate leakage density increases exponentially with the thickness of the gate dielectric films, which will result in the diffusion of foreign ions and generate unacceptable direct tunneling leakage currents, further resulting in the degradation of the device performance and seriously affecting the reliability of the MOS devices. To resolve this problem, high-$k$ dielectric materials have been introduced to replace SiO$_2$. Among them, HfO$_2$ has attracted much attention due to its excellent performance. HfO$_2$ has a high dielectric constant (~25), wide band gap (~5.8 eV), thermally stable contact with Si, and compatibility with a metal gate. However, HfO$_2$ also has many shortcomings, such as a lower crystallization temperature (600 °C), a large amount of negative charges in the film, and moderate dielectric constants, which limit further device scaling. Thus, the modification of HfO$_2$ is still a continuing area of research. In order to improve the performance of HfO$_2$, researchers have tried to dope other elements, such as Al, Ti, N, and Si, into HfO$_2$ to obtain high-$k$ gate dielectrics. It is found that the doping of new elements can only partially improve the characteristics of HfO$_2$. The introduction of new elements also introduced some new problems. For example, Al and Si incorporation can reduce the leakage current of HfO$_2$ but does not increase the dielectric constant. The addition of TiO$_2$ in HfO$_2$ can significantly increase the dielectric constant but leads to serious deterioration in the interface quality.

Presently, some researchers have also tried to incorporate two or more complementary elements into HfO$_2$ simultaneously. Li et al. successfully fabricated HfTTaO gate dielectrics by radio frequency (rf) sputter-deposition and physical vapor deposition (PVD) technology, respectively, which improved the performance by doping Ti and Ta into HfO$_2$. Lu et al. doped Ti and Al in the HfO$_2$ cubic structure on a Si substrate using sputtering technology and obtained an increase in the dielectric constant and an acceptable barrier height. These previous studies have focused mainly on the effect of the Ti content on the electrical properties of Hf-based MOS devices. However, few systematic studies have explored the effect of the doping ratio of Ti, Ta, or Al into HfO$_2$ on the relationship between the interfacial properties, band offset and the electrical characteristics. In addition, considering the decreasing trend in device size, PVD and sputtering technology cannot meet the demand for the
fabrication of ultrathin dielectric films for semiconductor devices. In comparison, atomic layer deposition (ALD) technology has outstanding advantages in obtaining high-κ films with high dielectric quality as well as accurate and controllable thicknesses. Thus, ALD has developed into a more popular technology for preparing ultrathin gate dielectric films in recent years. Based on the above analysis, in this work, ternary oxide films, HfTiO (HTO) and TiAlO (TAO), and a quaternary oxide film, HfTiAlO (HTAO), were prepared with different doping ratios of each element by ALD technology. The effect of the doping ratio on their interfacial properties, band offset and electrical characteristics, as well as the relationship between these properties were systematically investigated and analyzed.

2. Experimental

2.1 Preparation of samples

n-Doped Si (100) wafers with the resistance of 10–20 Ω were chosen as substrates. Before the deposition of the films, the Si wafers were cleaned by a standard RCA cleaning method. Laminate films, including HTO, TAO, and HTAO, were deposited by atomic layer deposition (ALD) technology in an ALD chamber (LabNano™ 9100, ENSURE NANOTECH). Tetra-kis(dimethylamino)hafnium (TDMAH), tetra-kis(dimethylamino)titanium (TDMAT), and trimethylaluminum (TMA) were chosen as the metal precursors for HfO2, TiO2, and Al2O3, respectively. H2O was used as the oxidant for all metal oxides. TDMAH and TDMAT were heated to 80 °C and H2O and TMA were kept at room temperature. All HTO, TAO, and HTAO films were obtained from ALD technology for preparing ultrathin gate dielectric films in recent years. The radiation source was equipped with an Al Kα radiation source (Eg = 1486.6 eV). All the XPS spectra were detected at a take-off angle of 45° and with an energy interval of 0.1 eV by means of a hemispherical analyzer and the pass energy was 20 eV. The absorption spectra of all films over the range of 300–900 nm were measured at room temperature using the SHIMADZU UV-2550 ultraviolet-visible spectrometer (UV-vis). The band gap values of films were obtained from the absorption spectrum by optical methods. The DC capacitance–voltage (C–V) and leakage current–voltage (I–V) characteristics of the MOS devices were measured using an Agilent B1500A semiconductor parameter analyzer combined with a Cascade Probe Station at room temperature and in a shielding dark case.

3. Results and discussion

3.1 HTO capacitors

3.1.1 Interface chemistry characterization. Fig. 2(a) shows the core-level XPS survey spectra for S1–S6. It is obvious that from S1 to S6, the intensity of the Hf peak gradually weakened and the intensity of each Ti peak was gradually enhanced with the change of the ALD cycle ratio of Hf:T. The changes in the interface chemical composition could be obtained by determining the Si 2p core-level spectra, which are fitted with seven peaks as exhibited in Fig. 2(b). The Si 2p doublet peaks (resulting from a spin–orbit splitting of 0.5 eV between 2p1/2 and 2p3/2) are located at 99.3 eV and 99.8 eV, which are fitted with a full width at half maximum (FWHM) of about 0.6 eV. The other five peaks related to the intrinsic oxides of Si (i.e., Si2+, Si3+, and Si4+) and silicate (Si−) were fit with a FWHM of about 1.15 eV and are located at 100.3 eV, 100.95 eV, 101.8 eV, 102.6 eV, 103.2 eV, arranged by binding energy from low to high, respectively. In order to obtain the change of chemical composition at the interfacial layer more clearly, the relative peak area percentages of silicate and four Si oxides were also calculated and are listed in Table 2. It can be seen that the relative peak area percentages of silicate in S1–S6 are 16.93%, 20.97%, 27.33%, 33.31%, 33.22% and 27.63%, respectively. It is observed that the silicate content for S1 in the interfacial layer between the HfO2 film and Si substrate is minimized. In contrast, for S6, the silicate composition in the interface layer between TiO2 film and Si substrate is higher. For S2–S5, the silicate component rises with increasing Ti content in the HTO films, which is consistent with Ye et al.Ye et al. found that the
interface quality would deteriorate with increasing Ti content, which is due to the increase of silicate. So, it can be summarized that TiO₂ can more easily react with Si than HfO₂ and form more silicate during deposition in the ALD chamber. The incorporation of Ti into HfO₂ is detrimental to the interface quality between the HTO film and Si wafer.

3.1.2 Band alignment. It is well known that the $k$ values of gate dielectrics are inversely related to the band gap ($E_g$) values. The higher the $k$ values are, the greater the physical thickness of the gate dielectric material is, so as to effectively inhibit the direct tunneling current. However, more often than not, the corresponding band gap value is smaller. So, the Schottky emission induced by thermonic emission usually takes place through the valence or conduction band edge. To limit the leakage current, the candidate high-$k$ dielectrics should satisfy the requirement that both the valence band offset (VBO) and the conduction band offset (CBO) are larger than 1 eV. So, the band alignment analysis determined by the energy of the VBO and CBO plays a critical role in the evaluation of high-$k$ material performance. The energy band gaps edges are measured through the UV-vis absorption spectrum and attained using the linear extrapolation method. In order to obtain the VBO values for high-$k$ gate dielectric films, the valence band offset ($\Delta E_{v}$) values are determined by Kraut’s method as shown in the relation below:

$$\Delta E_v(\text{high-}k-\text{Si}) = E_v(\text{high-}k) - E_v(\text{Si}) \quad (1)$$

The $E_v$ values are also attained using the linear extrapolation. At last, the conduction band offset ($\Delta E_{c}$) values can be obtained by the equation below:

Table 1  Summary of all samples with different precursor cycle ratio

| HfTiO Precursor ratio of Hf(x) : Ti(y) | TiAlO Precursor ratio of Ti(y) : Al(z) | HfTiAlO Precursor ratio of Hf(x) : Ti(y) : Al(z) |
|--------------------------------------|--------------------------------------|--------------------------------------|
| S1 1 : 0 | S1′ 0 : 1 | S1′′ 1 : 1 : 1 |
| S2 6 : 1 | S2′ 1 : 1 | S2′′ 2 : 1 : 1 |
| S3 4 : 1 | S3′ 2 : 1 | S3′′ 4 : 1 : 1 |
| S4 2 : 1 | S4′ 4 : 1 | S4′′ 6 : 1 : 1 |
| S5 1 : 1 | S5′ 6 : 1 | S5′′ 8 : 1 : 1 |
| S6 0 : 1 | S6′ 1 : 0 | |

Table 2  The relative peak area percentages of Si$^{1+}$, Si$^{2+}$, Si$^{3+}$, Si$^{4+}$, Si$^*$ in Si 2p peaks for S1–S6

| Si$^{1+}$ (%) | Si$^{2+}$ (%) | Si$^{3+}$ (%) | Si$^{4+}$ (%) | Si$^*$ (%) |
|-------------|-------------|-------------|-------------|---------|
| S1 23.06    | 14.89      | 38.95      | 6.16        | 16.93   |
| S2 20.18    | 8.15       | 41.02      | 9.68        | 20.97   |
| S3 10.12    | 10.63      | 42.63      | 9.29        | 27.33   |
| S4 15.84    | 6.38       | 28.52      | 15.95       | 33.31   |
| S5 17.22    | 9.09       | 32.56      | 9.16        | 33.22   |
| S6 13.33    | 10.92      | 37.46      | 10.63       | 27.63   |

Fig. 1  A diagram of the Si–MOS device.

Fig. 2  (a) Core level XPS spectra for all compositions and (b) Si 2p XPS spectra analysis of S1–S6.
Using the approach described above, the $E_g$ and $E_v$ for S1–S6 are depicted in Fig. 3(a) and (b). The $E_g$ of HfO$_2$ and TiO$_2$ is 5.93 eV and 3.66 eV, respectively. The $E_g$ of S2–S5, i.e., the HTO films, is found to be smaller than that of HfO$_2$ and larger than that of TiO$_2$, and decreases from 5.49 eV to 4.10 eV with increasing Ti content. The $\Delta E_c$ values of S1–S6 show the same trend, with the values of 2.39, 2.15, 2.01, 1.89, 1.81, and 1.56 eV, respectively. The decreased conduction band offsets with increasing Ti content are observed in Fig. 4. As displayed in Fig. 4, the extracted $\Delta E_c$ of S1–S6 are 2.42, 2.22, 2.14, 1.82, 1.17, and 1.02 eV, respectively. These are consistent with our previous research findings and Ye’s findings obtained by a magnetron sputtering method.$^{15,17}$

As reported in the previous literature, the $E_g$ of TiO$_2$ is dominated by Ti 3d and O 2s or O 2p states, as well as Hf 5d and O 2p states that contribute to the $E_g$ of HfTiO.$^{18,19}$ Because the energy of the Hf 5d state is above that of Ti 3d, the CB edge and the band gap of HfTiO are reduced with increasing Ti content.

### 3.1.3 Electrical characterization.

Double sweep C–$V$ measurements for all samples were taken at high frequency (1 MHz) along the clockwise direction. For high-$k$/Si stacks, there are four types of oxide charges on the side of the high-$k$ dielectrics. According to their distribution, they are also called fixed oxide charges, mobile ion charges, oxide trapped charges, and interface trapped charges, respectively. These defective charges in the oxide film will lead to the deviation between the measured CV curves in the experiment and the ideal CV curves.
Among them, the fixed oxide charges and the mobile ion charges will cause the CV curve to drift along the voltage axis; the oxide trapped charges can cause hysteresis of the CV curve; and the interface trapped charges will bring about CV distortion. According to the accumulated capacitance ($C_{\text{acc}}$), the corresponding permittivity ($k$) values can be extracted. From the deformation of the CV curves, the flat band voltage ($V_{\text{fb}}$) and hysteresis of the flat band voltage ($\Delta V_{\text{fb}}$) can be calculated. Furthermore, the fixed charge density ($Q_{\text{ox}}$), the trapped oxide charge density ($Q_{\text{bt}}$) can be obtained. So, from the CV curves, we can extract some electrical parameters to evaluate the quality of the high-$k$ dielectric thin films and the electrical properties of the MOS capacitors.

The bias voltage shift of the CV curve is mainly due to the fixed oxide charges and the native defect charges in the oxides near the interface region. When the fixed oxide charges are positive, the corresponding CV curve moves in the direction of the negative voltage; whereas if the fixed oxide charges are negative, the CV curve will move toward the positive voltage. $Q_{\text{ox}}$ can be obtained using the following formula:\cite{20, 21}

$$Q_{\text{ox}} = -C_{\text{acc}}(V_{\text{fb}} - \phi_{\text{ms}})/Aq$$  \hspace{1cm} (3)

in which $\phi_{\text{ms}}$ is the difference in work function between the aluminum electrode and Si substrate, $A$ is the aluminum metal electrode area, and $\Delta V_{\text{fb}}$, depending on the bias, is ascribed to the trapped charges existing in the oxide and interface layer. These trapped oxide charges include the hole trapping state and the electron trapping state, which can trap holes and electrons in the oxide film, respectively, and make them charged positively and negatively. Thus, in the process of a bi-directional sweep voltage, the bias voltage hysteresis of the CV curve occurred. So, the density of trapped oxide charges can be obtained by the formula:\cite{20, 22}

$$Q_{\text{bt}} = C_{\text{acc}}\Delta V_{\text{fb}}/Aq.$$  \hspace{1cm} (4)

Fig. 5(a) shows the C–V curves of Al/HTO/Si–MOS capacitors with different Ti doping content. It can be seen that the $C_{\text{acc}}$ of S1, i.e., HfO$_2$, is the minimum, whereas, that of S6, i.e., TiO$_2$, is the maximum. As for HTO, the $C_{\text{acc}}$ values are larger than that of HfO$_2$ and smaller than that of TiO$_2$, and rise with increasing Ti content. The corresponding $k$ values extracted from $C_{\text{acc}}$ are 13.9, 16.4, 19.6, 22.1, 23.1, and 29.5, respectively. The $k$ values increase with increasing Ti content and are consistent with those reported in the literature.\cite{15, 17} For S1, there is a left flatland shift in the CV curve, which is due to the positive trapped oxide

| Composition | $C_{\text{acc}}$ (F) | $\epsilon_{\text{dr}}$ | $C_{\text{bt}}$ (pF) | $V_{\text{fb}}$ (V) | $\Delta V_{\text{fb}}$ (V) | $Q_{\text{ox}}$ (cm$^{-2}$) | $Q_{\text{bt}}$ (cm$^{-2}$) | $J$ (A cm$^{-2}$) |
|-------------|----------------------|----------------------|----------------------|----------------------|----------------------|----------------------|----------------------|----------------------|
| S1          | $2.39 \times 10^{-10}$ | 13.9                 | $1.30 \times 10^{-10}$ | 0.11                 | 0.09                 | $2.08 \times 10^{12}$ | $-5.51 \times 10^{12}$ | $2.07 \times 10^{-5}$ |
| S2          | $4.11 \times 10^{-10}$ | 16.4                 | $1.65 \times 10^{-10}$ | -0.54                | 0.08                 | $1.74 \times 10^{12}$ | $-4.73 \times 10^{12}$ | $3.06 \times 10^{-5}$ |
| S3          | $4.73 \times 10^{-10}$ | 19.6                 | $1.65 \times 10^{-10}$ | -0.07                | —                    | $-5.48 \times 10^{11}$ | —                    | $1.11 \times 10^{-4}$ |
| S4          | $5.19 \times 10^{-10}$ | 22.1                 | $1.70 \times 10^{-10}$ | 0.01                 | —                    | $-3.79 \times 10^{11}$ | —                    | $2.65 \times 10^{-5}$ |
| S5          | $5.37 \times 10^{-10}$ | 23.1                 | $1.73 \times 10^{-10}$ | 0.07                 | —                    | $-7.22 \times 10^{11}$ | —                    | $1.28 \times 10^{-4}$ |
| S6          | $6.41 \times 10^{-10}$ | 29.5                 | $1.90 \times 10^{-10}$ | 0.255                | —                    | $-1.84 \times 10^{12}$ | —                    | $1.53 \times 10^{-2}$ |

| Composition | $\text{Si}^{1+}$ (%) | $\text{Si}^{2+}$ (%) | $\text{Si}^{3+}$ (%) | $\text{Si}^{4+}$ (%) | $\text{Si}^{*}$ (%) |
|-------------|----------------------|----------------------|----------------------|----------------------|----------------------|
| S1’         | 43.37                | 22.05                | 21.69                | 4.99                 | 7.90                 |
| S2’         | 6.00                 | 14.31                | 51.00                | 12.06                | 16.63                |
| S3’         | 20.61                | 13.88                | 39.52                | 4.93                 | 21.06                |
| S4’         | 16.62                | 6.76                 | 41.38                | 8.36                 | 26.88                |
| S5’         | 11.81                | 5.71                 | 46.17                | 9.60                 | 26.71                |
| S6’         | 13.35                | 10.92                | 37.46                | 10.64                | 27.63                |

Fig. 6 (a) Core level XPS spectra for all compositions and (b) Si 2p XPS spectra analysis of S1’–S6’.
charges or positive intrinsic defects existing in the HfO$_2$ film and/or in the HfO$_2$/Si interface layer.$^{20,22}$ For S6, a right flatland shift in the CV curve is observed. On one hand, it is because of the difference in work function between the gate dielectric film and Si substrate; on the other hand, it is due to the negative trapped oxide charges or negative intrinsic defects existing in the TiO$_2$ film and/or in the TiO$_2$/Si interface layer.$^{20,22}$ As for S2–S5, after doping TiO$_2$ into HfO$_2$, there is a great negative shift in the $V_{th}$ as compared to HfO$_2$ and TiO$_2$. The $Q_{ox}$ values of all samples can be calculated and is listed in Table 3 they are $2.08 \times 10^{12}$, $1.74 \times 10^{13}$, $-5.48 \times 10^{13}$, $-3.79 \times 10^{11}$, $-7.22 \times 10^{11}$, $-1.84 \times 10^{12}$ cm$^{-2}$, respectively. It can be observed that the $Q_{ox}$ values for the HTO samples vary from positive to negative with increasing Ti content. It may be due to the increase of the positive trapped oxide charges or negative intrinsic defects. These charges may originate from the oxygen vacancies in oxides caused by dangling bonds on the Si surface, which originates from the lack of oxygen atoms during the gate dielectric film deposition.$^{20,22}$ A flatland shift along the positive voltage direction is observed with increasing Ti content, which is attributed to an increase in the positive oxide charges in the films near the interface layers.$^{20,22}$

Additionally, it can be observed that there is obvious hysteresis and stretching in the depletion region for S1; while for S6, there is almost no hysteresis and no stretching in the depletion region. Compared with S1 and S6, the CV curves of S2–S5 changed significantly after doping TiO$_2$ into HfO$_2$ to form HTO samples. When the Hf content is higher, such as in S2, the $\Delta V_{th}$ increases and the stretching in the depletion region and distortion also worsen. Interestingly, as the content of Ti...
increases gradually, the voltage hysteresis is quickly eliminated. The density of border trapped charges causing hysteresis is calculated and listed in Table 3. The \( Q_{\text{bt}} \) for HfO\(_2\) (S1) is \(-5.51 \times 10^{11}\) cm\(^{-2}\) and that for TiO\(_2\) (S6) is almost 0. Also, the \( Q_{\text{bt}} \) for other HTO are 0, except for S2, which corresponds to the maximum HF content.

As depicted in Fig. 5(b), the leakage current densities from the gate voltage \((J-V)\) curves for all samples were measured by sweeping the voltage from \(-3\) V to \(+3\) V, and the corresponding leakage current densities \(J\) at \(+1\) V were extracted to compare. As listed in Table 3, the \( J \) values of S1–S6 are \(2.07 \times 10^{-3}, 3.06 \times 10^{-5}, 1.11 \times 10^{-2}, 2.65 \times 10^{-3}, 1.28 \times 10^{-2}, \) and \(1.53 \times 10^{-2}\) A cm\(^{-2}\), respectively. Combined with XPS fitting results, band alignment and electrical analysis, this indicates that S5 (ALD cycle ratio of HF : Ti = 1 : 1) presents ideal CV characteristics with the smallest flat voltage and almost no hysteresis, and possesses excellent film properties with the least defect charges. However, the poor interface properties and smaller conduction band offset inhibit the improvement of the electrical properties. S5 has a modest dielectric constant \((-23.1)\), but the leakage current \((1.28 \times 10^{-4}\) A cm\(^{-2}\)) is also larger. It is obvious that doping Ti into HfO\(_2\) cannot simultaneously improve both: the cost of improving the dielectric constant is an increase in the leakage current.

3.2 TAO capacitors

Based on the interface analysis for HTO, it can be demonstrated that TiO\(_2\) is more susceptible to reaction on a Si substrate and forms more silicate during deposition as compared with HfO\(_2\). To inhibit the diffusion between TiO\(_2\) and the Si substrate and suppress the leakage current, Al\(_2\)O\(_3\), which possesses good passivation properties, was doped into TiO\(_2\) to form a TAO laminated film.

3.2.1 Interface characterization. In Fig. 6(a), the survey spectra for S1–S6′ display the intensity of Ti peaks decreasing and the intensity of Al peaks increasing with the change of the ALD cycle ratio of Ti : Al from 0 : 1 to 1 : 0. Fig. 6(b) shows the Si 2p XPS core level of TAO/SiO. Similarly to the HTO XPS analysis above, Si–O bonds are deconvoluted into four oxides and silicate peaks. The relative peak area percentages of silicate and the four Si oxides are further calculated and listed in Table 4. It can be seen that the relative peak area percentages of silicate for S1′–S6′ are 7.90%, 16.63%, 21.06%, 26.88%, 26.71% and 27.63%, respectively. It is obvious that for S1′, there is only a small amount of silicate at the interface layer. By contrast, for S6′, there is a lot of silicate in the interface layer. For S2′–S5′, the silicate in the interface layer decreases with increasing Al content in TAO. It is not difficult to deduce that the incorporation of Al\(_2\)O\(_3\), which has excellent chemical stability and passivation effects, effectively suppresses the oxygen diffusion

Table 5  The electric parameters extracted for S1′–S6′

| TiAO   | \( C_{\text{max}} \) (pF) | \( \epsilon_{\text{bt}} \) | \( C_{\text{fb}} \) (pF) | \( V_{\text{fb}} \) (V) | \( \Delta V_{\text{fb}} \) (V) | \( Q_{\text{ox}} \) (cm\(^{-2}\)) | \( N_{\text{ox}} \) (cm\(^{-2}\)) | \( J \) (A cm\(^{-2}\)) |
|--------|--------------------------|------------------|-------------------|----------------|-----------------|-----------------|-----------------|-------------------|
| S1′    | \(2.21 \times 10^{-10}\) | 7.2              | \(1.31 \times 10^{-10}\) | \(-0.64\)      | 0.08            | \(-3.22 \times 10^{11}\) | \(6.03 \times 10^{-7}\) |
| S2′    | \(4.57 \times 10^{-10}\) | 18.4             | \(2.46 \times 10^{-10}\) | 0.076          | 0.066           | \(-7.72 \times 10^{11}\) | \(7.31 \times 10^{-6}\) |
| S3′    | \(4.76 \times 10^{-10}\) | 19.7             | \(2.14 \times 10^{-10}\) | 0.12           | 0.093           | \(-6.71 \times 10^{12}\) | \(2.39 \times 10^{-5}\) |
| S4′    | \(5.33 \times 10^{-10}\) | 22.9             | \(2.09 \times 10^{-10}\) | 0.32           | 0.085           | \(-4.41 \times 10^{12}\) | \(1.23 \times 10^{-4}\) |
| S5′    | \(7.05 \times 10^{-10}\) | 33.8             | \(1.86 \times 10^{-10}\) | \(-0.065\)     | 0.095           | \(-4.36 \times 10^{12}\) | \(4.02 \times 10^{-4}\) |
| S6′    | \(6.50 \times 10^{-10}\) | 29.5             | \(1.90 \times 10^{-10}\) | 0.255          | 0.01            | \(-4.15 \times 10^{12}\) | \(1.53 \times 10^{-2}\) |

Table 6  The relative peak area percentages of Si\(^{1+}\), Si\(^{2+}\), Si\(^{3+}\), Si\(^{4+}\), Si\(^{\ast}\) in Si 2p peaks for S1′–S6′

| Si\(^{1+}\) (%) | Si\(^{2+}\) (%) | Si\(^{3+}\) (%) | Si\(^{4+}\) (%) | Si\(^{\ast}\) (%) |
|----------------|----------------|----------------|----------------|----------------|
| S1′            | 14.85          | 9.44           | 39.53          | 9.11           | 27.07          |
| S2′            | 15.93          | 10.77          | 40.42          | 12.93          | 19.95          |
| S3′            | 11.04          | 7.68           | 56.33          | 11.14          | 13.81          |
| S4′            | 12.75          | 10.52          | 57.66          | 8.60           | 10.47          |
| S5′            | 11.11          | 11.11          | 43.07          | 11.68          | 23.03          |

Fig. 10  (a) Core level XPS spectra for all compositions and (b) Si 2p XPS spectra analysis of S1′–S6′.
downward into the Si substrate, thus reducing the generation of silicate. This is in agreement with our previous research on HAO/Si gate stacks.

3.2.2 Band alignment. The band alignment of the TAO film with the Si substrate was determined using the method described above for the analysis of the HTO film. Fig. 7(a) shows the experimental data on the band gaps for S1–S6. The $E_g$ estimated for pure $\text{Al}_2\text{O}_3$ and $\text{TiO}_2$ are about 5.99 eV and 3.73 eV, respectively. The $E_g$ of the TAO film gradually increases as the Ti content decreases in the range of 3.86–4.31 eV. The results are in agreement with the values reported in previous studies. The $\Delta E_v$ for S1–S6 as a function of Ti content are shown in Fig. 7(b). This reveals that there is a gradual increase in $\Delta E_v$ with decreasing Ti content. The $\Delta E_v$ value of TAO is closer to that of $\text{TiO}_2$ and seems to be dominated by that of $\text{TiO}_2$, which was observed by An et al. and Alekhin et al. for TAO on GaAs and Si, respectively. Using formula (2) above, the $\Delta E_c$ for S1–S6 are calculated and presented in Fig. 8. It is well known that the conduction band offset of the $\text{TiO}_2$/Si heterojunction is below 1 eV. In this work, the $\Delta E_c$ obtained for $\text{TiO}_2$ is 0.93 eV, which is consistent with data obtained in earlier studies. The $\Delta E_c$ for TAO slowly increases from 1.01 eV to 1.19 eV with the increase of Al content and is closer that of $\text{TiO}_2$. A possible explanation for these results may be that the Al content in the TAO cannot effectively increase the $\Delta E_c$ values relative to n-Si. These results can be attributed to the interfacial layer between TAO and the Si substrate that tunes the $\Delta E_c$ values, as reported in the studies of An et al. and Perego et al.

3.2.3 Electrical properties. The electrical properties of Al/TAO/Si stacks are characterized by $C$–$V$ and $J$–$V$ measurements as exhibited in Fig. 9(a) and (b). It can be seen that the
incorporation of Al₂O₃ modulates the electrical properties of TiO₂. The $C_{\text{acc}}$ of S₂’–S₅’ first increases and then decreases with increasing Al content. The change of $C_{\text{acc}}$ can be explained by a small amount of Al doping that improves the interfacial properties between TAO and the Si substrate and inhibits the formation of silicates, and thus enhances $C_{\text{acc}}$; however, with the increase of Al content, the effect of Al₂O₃ with low permittivity on TAO is more and more significant, which will lead to the decrease of $C_{\text{acc}}$. The permittivity of the TAO film is calculated from $C_{\text{acc}}$ and listed in Table 5. From the bi-directional CV curves, it can be seen that the $\Delta V_{\text{th}}$ increases obviously after the incorporation of Al into TiO₂. Taking into account the postulation mentioned in the case of HTO, it can be speculated that the trapped charges existing in the TiO₂/Al₂O₃ laminate film lead to the increase in $\Delta V_{\text{th}}$. Meanwhile, the shifts of the TAO $V_{\text{th}}$ relative to TiO₂ are due to the change in the oxide charges in the TAO film near the interface layers. It can be observed that the CV curves move first to the right and then to the left, which indicates that the fixed charges in the TAO film increase first and then decrease. Additionally, with the increasing Al content in the TAO film, the upward trend in the accumulation area of the CV curves was eliminated, which symbolizes that the interface state is suppressed. From Fig. 9(b), it can be clearly seen that there was a marked decrease in the gate leakage current with the increase of Al content in the TAO film. Additionally, S₂’ owns the lowest leakage current density (~7.31 × 10⁻⁶ A cm⁻²).

Combining the analysis results extracted from XPS, band alignment, and electrical characteristics, it can be concluded that the incorporation of Al₂O₃ in TiO₂ can reduce the generation of SiO₂ and silicate, improve the quality of the interface and TAO films, increase $E_g$ and $\Delta E_C$ relative to Si, and reduce the gate leakage current, but at the expense of permittivity. In this work, S₂’ (precursor ratio of T : Al = 1 : 1) exhibits excellent interface and film properties, the smallest leakage current (~7.31 × 10⁻⁶ A cm⁻²), but has a low $k$ value (~18.4). So, when Al is used to modulate the properties of the TiO₂/Si stack, the doping amount of Al should be strictly controlled.

### 3.3 HTAO capacitors

The evidence presented in the preceding two sections suggests that adding TiO₂ into HfO₂ can increase the permittivity, but...
deteriorate the interface; while incorporating Al₂O₃ into TiO₂ can improve the quality of the interface and limit the leakage current of films, but it must be at the expense of decreasing the dielectric constant. In our previous work, HAO obtained through doping Al₂O₃ into HfO₂ had the same problems as TAO.⁷,₂⁶,₂⁷ So, in this section, both Al₂O₃ and TiO₂ are incorporated into HfO₂ to form a quaternary oxide, HTAO. Based on the analysis above for the TAO/Si stacks, the sample S₂ with the precursor ratio of T : Al = 1 : 1 has excellent characteristics in the film and interface layer, and the smallest leakage current. So, in this work, the precursor ratio of T : Al was fixed at 1 : 1 and the number of cycles of Hf precursor was changed to obtain HTAO with a different doping ratio.

### 3.3.1 Interface characterization

Fig. 10(a) shows the survey spectra for S₁⁰⁰–S₅⁰⁰. It can be seen that the intensity of Al 2s and Ti 2p decrease gradually from S₁⁰⁰ to S₅⁰⁰ with increases in the ALD precursor cycle number of HfO₂. Fig. 10(b) exhibits the Si 2p XPS spectra of HTAO/Si. Si 2p is deconvoluted into seven peaks following the spectra-decomposition method used in Section 3.1. What can be clearly seen in Fig. 10(b) is the changes of silicate and silica oxide content with the incorporation of Ti and Al into HfO₂. The relative peak area percentages of silicate and the four Si oxides for S₁⁰⁰–S₅⁰⁰ are also calculated and listed in Table 6. It can be seen that the relative peak area percentages of silicate are 27.07%, 19.95%, 13.81%, 10.47%, and 23.03%, respectively. With the decrease of Ti and Al content, i.e., the increase of Hf content in HTAO, the silicate at the interfacial

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**Fig. 15** (a) SE emission, (b) PF emission and (c) FN tunneling plots for S₄⁰⁰ MOS capacitor under the gate injection.

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**Fig. 16** The energy band structure for the Al–HTAO–Si–MOS capacitor under the gate injection.
layer decreased first and then increased. When the ALD cyclic ratio of Hf:Al is 1:6:1, the silicate is the lowest. Combining the XPS analysis results in Section 3.1, TiO₂ diffuses more readily into Si than HfO₂. When the ALD cyclic ratio of Ti:Al is fixed at 1:1, the increasing Hf content would inhibit the diffusion of Hf and Ti into the Si substrate. Therefore, as the ALD cyclic ratio of Hf:Al increased to 8:1:1, the silicate and SiO₂ in the interface layer increased instead.

3.3.2 Band alignment. The band alignment of the HTAO film with Si substrates was determined using the method described in Section 3.1. In Fig. 11(a), the $E_g$ of S1â”“S5" are 5.82, 5.72, 5.60, 5.16, and 4.79 eV, respectively. The $E_g$ values of HTAO are smaller than those of Al₂O₃ but higher than TiO₂, and closer to HfO₂ with increasing Hf content. These results are consistent with those reported in Hf₀.5₂Ti₀.3₃Al₀.₁₅O by Aleklin et al. 32. The $\Delta E_s$ for S1â”“S5" as a function of Hf content are shown in Fig. 11(b). It is apparent that there is a gradual increase from 1.93 to 2.19 eV with the increase in Hf content. Following formula (2) in Section 3.1, the calculated $E_{tc}$ for S1â”“S5" are exhibited in Fig. 12. The CBO of HTAO/Si increases from 1.74 to 2.51 eV with the increase of Hf content. In addition, it can be seen that there was a marked increase in the CBO as the ALD cycle of the Hf precursor varies from 1 to 4; while as the ALD cycle of the Hf precursor varies from 4 to 8, the CBO exhibits a slight increase. Compared with the above XPS analysis of HTO/Si and TAO/Si, it is possible conclude that the performance of the HfO₂/Si interface is better than that of the TiO₂/Si interface. Thus, the appropriate increase of Hf content in the HTAO film can improve the interfacial properties. Nevertheless, with the continued increase of Hf content, the decrease in the relative composition of Al is not sufficient to inhibit the diffusion between Si and the HTAO film, which will lead to the deterioration of interface performance.

3.3.3 Electrical properties. In Fig. 13(a), the C-V curves for the Al/HTAO/Si-MOS capacitors are presented. Compared with the HTO and TAO samples, it is obvious that the shape of the C-V curves for HTAO is closer to that of an ideal C-V curve. The voltage hysteresis of all curves is very small and the values are very close around 0.03 V. Such a small hysteresis indicates that the trapped charges existing in the HTAO films and interface layers are reduced by doping Hf, Ti, and Al oxides together. From Table 7, the extracted $N_{ox}$ for S1â”“S5" is $-1.44 \times 10^{11}$, $-1.59 \times 10^{11}$, $-2.10 \times 10^{11}$, $-3.06 \times 10^{11}$, and $-3.48 \times 10^{11}$ cm$^{-2}$, respectively. The number of trapped charges in the quaternary oxide HTAO is one order of magnitude lower than that of the ternary oxides HTO and TAO. From our experimental results, it can be seen that for TiO₂, there is almost no hysteresis in the MOS devices, so trapped charges in the TiO₂ film can be ignored. Additionally, Al₂O₃ plays a passivation role to improve the quality of the interface layer. Therefore, doping Al₂O₃ and TiO₂ into HfO₂ can not only improve the interface quality, but also reduce the voltage hysteresis, thus reducing the number of trapped charges in the HTAO film.

The deviation of the CV curve to a more positive voltage decreases with the increase of Hf content. The corresponding $V_{fb}$ for S1â”“S5" are 0.33, 0.21, 0.083, 0.200, and −0.069 V, respectively. The deviation of the CV curves is mainly due to the negative fixed oxide charges and the native defect charges in the HTAO films that decrease with increasing Hf content. The extracted $Q_{ox}$ for S1â”“S5" are $-2.25 \times 10^{12}$, $-1.92 \times 10^{12}$, $-1.25 \times 10^{12}$, $-7.75 \times 10^{11}$, and $-9.33 \times 10^{10}$ cm$^{-2}$, respectively. With the increase of Hf content, the number of oxide charges in S5" is about two orders of magnitude lower than that of S1". Moreover, $C_{acc}$ increases obviously with increasing Hf content too. The permittivity for S1"−S5" increases in turn and are 12.8, 17.7, 23.9, 28.8, and 35.5, respectively. These results are in good agreement with the experimental studies from Lu et al. and Alekhin et al. Combined with the preceding analysis, the relative content of Hf and Ti with a high dielectric constant increases; on the other hand, the reduction of SiO₂ and silicate with a low dielectric constant is constant at the interface layer. Simultaneously, the CV curve for S1" has evident unsaturation in the accumulated region, indicating that the high interface states exist in the HTAO/Si interface. With increasing Hf content, the accumulation region of the CV curve shows a trend of saturation at first as the number of Hf cycles rises from 1 to 6, and then the unsaturated phenomenon appears again when the number of Hf cycles is 8. The results may be due to the fact that the appropriate increase of relative content of Hf in the HTAO film can reduce the interface states in the HTAO/Si interface layer to improve the interfacial properties. This is consistent with the results of the XPS analysis above. Fig. 13(b) shows the J-V characteristics for Al/HTAO/Si-MOS capacitors under the gate and substrate injection mode. Compared with the HTO and TAO samples, it is notable that HTAO samples exhibit a relatively narrow range of lower leakage current from 2.76 $\times$ $10^{-4}$ A cm$^{-2}$ to 1.11 $\times$ $10^{-3}$ A cm$^{-2}$ at 1 V. The leakage current decreases first and then increases with increasing Hf content. When the ALD cyclic ratio of Hf:Ti:Al equal to 6:1:1, the leakage current density is smallest ($\sim$1.11 $\times$ $10^{-3}$ A cm$^{-2}$).

Combining the results extracted from XPS analysis, band alignment, and electrical measurements, it can be concluded that the Hf, Ti, and Al elements are doped in a certain proportion to form a quaternary oxide (HTAO) with excellent properties.

3.4 Mechanism of carrier charge transport (MCCT) The phenomenon of carrier injection into the gate oxide by tunneling is due to the quantum-mechanical nature of the electron. The MCCT is very complex. For a specific gate dielectric, one or more MCCT plays a major role in a given voltage range. To study the mechanism of carrier charge transport in the Al/HTAO/Si-MOS capacitor, several models for the S4" with the best performance were discussed and evaluated in detail for both the gate injection and substrate, including Schottky emission (SE), Poole–Frenkel (PF) emission, and Fowler–Nordheim (FN) tunneling. As shown in Fig. 14 and 15, it is found that the dominating MCCT follows SE emission in the lower electric
field, PF emission in the medium electric field, and FN tunneling in the higher electric field.

The SE emission model can be described by the equation:

$$J_{SE} = A^* T^2 \exp \left[ -q \left( \phi_B - \sqrt{qE/4\pi\varepsilon_0\varepsilon_{ox}} \right) \right]$$

in which $A^* = 120(m_ox^*/m_0)$ (A cm$^{-2}$ K$^{-2}$), $T$, $q$, $k_B$, and $E$ denote the Richardson constant, the absolute temperature, the electronic charge, and the Boltzmann constant, respectively. $\varepsilon_0$ and $\varepsilon_{ox}$ are the permittivity of vacuum and the relative dielectric constant, respectively. $\phi_B$ represents the potential barrier height between the Al electrode and the conduction band edge $E_c$ of HTAO. As shown in Fig. 14(a) and 15(a), there is a linear relationship between $\ln(J/E^2)$ and $E^{1/2}$ (about 6.25–9.61 MV cm$^{-1}$ and 2.56–6.76 MV cm$^{-1}$ at substrate injection and gate injection mode), which indicates that the SE emission occurs through the HTAO film. From the extracted equation: slope $= 1/kT \sqrt{q^3/4\pi\varepsilon_0\varepsilon_{ox}}$, the $\varepsilon_{ox}$ and $n$ can be obtained. For the substrate injection and gate injection, the $\varepsilon_{ox}$ is 15.74 and 23.98, respectively. The $\varepsilon_{ox}$ values of gate injection are closer to the values extracted from the CV curve. This result indicates that the SE emission is the dominant current transmission mechanism in the gate injection mode. The corresponding intercept $= \ln(120(m_ox^*/m_0)) - \phi_B/kT$. Combined with the slope of the linear fitting line of FN MCCT, the $m_ox^*$ and $\phi_B$ can also be extracted.

The PF emission model follows the equation:

$$J_{PF} = (q\mu N_c)E \exp \left[ -q(\phi_c - \sqrt{qE/\pi\varepsilon_0\varepsilon_{ox}}) \right]$$

$\mu$ and $N_c$ are the mobility of the carrier and the carrier density, respectively. $\phi_c$ is the energy barrier from the traps to the conduction band of the dielectric. $\varepsilon_{ox}$ is the dynamic dielectric constant. $\varepsilon_{ox}$ is the square of the refractive index ($n$). For the PF model, $\ln(J/E)$ should have a linear relationship with $E^{1/2}$. Fig. 14(b) and 15(b) show $\ln(J/E)$ vs. $E^{1/2}$. The plots indicate that there is a linear relation in the middle electric field region (about 6.25–20.24 MV cm$^{-1}$ and 1.44–7.84 MV cm$^{-1}$ at the substrate injection and gate injection mode). From the slope of the linear portion:

slope $= \sqrt{q^3/\pi\varepsilon_0}$, the $\varepsilon_{ox}$ values extracted are 1942.93 and 5.83 in the substrate injection and gate injection, respectively. The corresponding refractive index $n$ values are 44.08 and 2.42. Compared with $n$ values of Hf-based high-$k$ dielectrics reported in He et al.’s previous work$^{5,4}$, the $n$ value for the substrate injection is inconsistent with the experimental value, which indicates that PF emission is the dominant current transmission mechanism. From the intercept of the linear fitting line at the longitudinal coordinate: intercept $= \ln(C) - \phi_c/kT$, the $\phi_c$ value obtained for substrate injection is 1.90 eV.

For the FN tunneling model,

$$\ln(J_{FN}) = \frac{q^3E^2}{16\pi^2\hbar\varepsilon_{ox}} \exp \left[ -\frac{4\sqrt{2m^*_0}q\phi_B}{3\hbar qE} \right]$$

in which $m_0^*$ is the tunneling effective mass in the HTAO film. Under a high electric field, the carriers will tunnel through the insulator, i.e., electrons are injected into the conduction band of the oxide through the triangular barrier. In the FN model, $\ln(J/E^2)$ and $1/E$ would conform to the linear relationship. Fig. 14(c) and 15(c) show $\ln(J/E^2)$ versus $1/E$ under substrate injection and gate injection, respectively. It can be seen that the MCCT at a higher electric region appears to follow the FN model in both the substrate and gate injection mode. The slope of the linear portion corresponds to the following expression:

slope $= -6.83 \times 10^7 (m_ox^*/m_0)q\phi_B^2$. In conjunction with the intercept equation of the SE emission, $m_ox^*$ and $\phi_B$ can be obtained. Using a self-consistent method, through matching the intercept of the SE model and the slope of the FN model in gate injection mode, the functional relationship between $m_ox^*$ and $q\phi_B$ can be expressed as in Fig. 16. The red curve belongs to the FN mode and the blue curve corresponds to the SE mode. It can be seen that the $q\phi_B$ increases as the $m_ox^*$ decreases for the SE model, while for the FN model, the $q\phi_B$ increases as the $m_ox^*$ increases. The two curves intersect at a point whose value (0.33$m_0$, 1.2 eV) is the required $m_ox^*$ and $q\phi_B$ value. It is known that the work function of Al is 4.28 eV$^{23}$, the electron affinity potential of Si ($\chi$) is 4.13 eV, and the $E_c$ value of the HTAO film relative to Si is 2.47 eV as obtained above; thus, the energy band structure for the Al/HTAO/Si–MOS capacitor under gate injection is outlined in Fig. 17.

4. Conclusion

In this work, the interfacial characteristics, band alignment and electrical performance of ALD-grown HTO, TAO and HTAO gate dielectric films on n-Si wafers were investigated in detail. The experimental results indicate that HfO$_2$ doped with TiO$_2$ can increase the permittivity, but it deteriorates the interface quality and induces increased leakage currents. While incorporating Al$_2$O$_3$ into TiO$_2$ can improve the interface quality and reduce leakage current, it must be at the expense of the dielectric constant. Importantly, the quaternary oxide HTAO synthesized by doping TiO$_2$ and Al$_2$O$_3$ into HfO$_2$ has excellent interface quality and electrical properties. However, an appropriate element ratio is necessary for the HTAO film to possess excellent performance. The interface analysis, band alignment and electrical characterization indicate that the HTAO sample with an ALD cycle ratio of Hf : Ti : Al = 6 : 1 : 1 exhibits the most excellent film quality, interface performance and electrical properties.

Conflicts of interest

The authors declare no conflict of interest statement.

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