Large-Scale Bose-Einstein Condensation in an Atomic Gas by Applying an Electric Field

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Abstract: Large-scale Bose-Einstein condensation (BEC) of cesium atoms has been observed (T=343K). The technical bottleneck of BEC is very small trapping volume (10^-8 cm^3), which made the number of condensed atoms still stagnant (less than 10^7), much smaller than normal condensation (more than 10^13), large-scale BEC has never been observed. In BEC experiment, scientists have applied magnetic field (used to trap atoms) and laser (used to cool atoms), but never considered applying electric field, because they think that all kinds of atoms are non-polar atoms. The breakthrough of the bottleneck lies in the application of electric field. In theory, despite 6s and 6p states of cesium are not degenerate, but Cs may be polar atom doesn't conflict with quantum mechanics because it is hydrogen-like atom. When an electric field was applied, Cs atoms become dipoles, therefore large-scale BEC can be observed. BEC experiment of cesium has been redone. From the entropy S=0, critical voltage V_c=78V. When V < V_c, S > 0; when V > V_c, S<0, phase transition occurred. When V=370V, condensates contained up to 2.71×10^{17} atoms, a forty percent improvement over previous results. This BEC is a second-order phase transition because entropy doesn't show discontinuity, from Wikipedia. It is also equivalent to a quantum phase transition. The permanent dipole moment (PDM) of Cs atom has been accurately measured: d_{PDM}=[1.84±0.15(stat) ±0.11 (syst)] ×10^{-8} e.cm. The PDM doesn't arise from the nuclear spin but arises from asymmetrical charge distribution, and it doesn't violate both time reversal and parity symmetry. The maximum induced dipole moment is d_{ind} ≤ 2.12 ×10^{-13}e.cm, which can be ignored. BEC has three main features: BEC is macroscopic occupation of the ground state of the system; BEC is condensation in momentum space; Bose gas would undergo a phase transition. Our experiment is an ideal BEC because it conforms to the three features. BEC belongs to the condensation of quantum gas, when V > V_c, almost all Cs atoms (bosons) are in exactly the same state, according to Feynman, "the quantum physics is the same thing as the classical physics", so our classical theory can explain BEC experiment satisfactorily. Ultra-low temperature is to make Bose gas phase transition, we used critical voltage to achieve phase transition, ultra-low temperature is no longer necessary. Five innovative formulas were first reported in the history of physics, the publication of this article marking mankind will enter a new era of polar atoms.

Keywords: Presence of Polar Atom, Entropy of Bose Gas, Order Parameter of BEC, BEC as Second-Order Phase Transition, BEC as Quantum Phase Transition, Three Main Features of BEC, Non-zero PDM of Cs Atom

1. Introduction

Bose-Einstein condensation (BEC) is a very active research field in condensed matter physics and material science. A suitable example of BEC is provided by a gas of an appropriate density of alkali-metal atoms (density range from 10^{13} to 10^{19} cm^{-3}). All stable alkali species--- Li (1), Na (2), K (3), Rb (4), and Cs (5) --- have been condensed. In the past, there are two ways to achieve phase transition: given atomic density, lower the temperature of Bose gas, making T < T_c; or given temperature, increase the density of Bose gas, making n > n_c (the critical density n_c = \frac{2\pi\hbar^2}{mk^2} (\frac{2.612}{\sqrt{\frac{n}{2.612}}})^\frac{3}{2}. But the latter method is not successful, because the density of Bose gas can’t be increased indefinitely, and therefore the laser cooling is now widely used. The key requirement is ultra-low temperatures--less than a millithousand of a kelvin above absolute zero. Unlike the liquid-gas transition,
this transition is not driven by interactions between particles [6]. Although many advanced technologies are used, but the vast majority of atoms are randomly oriented, large-scale BEC has never been observed [1-6]. In November 1995, W. Ketterle used a cloud of sodium to create BEC with $5 \times 10^3$ atoms and won the Nobel Prize in Physics in 2001. He pointed out that “Optical dipole traps provide tight confinement, but have only a very small trapping volume ($10^{-5}$ cm$^3$).” [2] Eight years later, in the BEC experiment of cesium atoms, scientists created BEC with $1.6 \times 10^8$ atoms [5]. The trapping volume not only did not increase, but decreased to $0.5 \times 10^{-5}$ cm$^3$, because the density of cesium reached $1.3 \times 10^{13}$ cm$^{-3}$, the number of trapping atoms is only $6.5 \times 10^7$ [5]. This fact shows that the current technical bottleneck is a very small trapping volume ($10^{-5}$ cm$^3$). So the number of condensed atoms is still less than $10^7$ ($10^{13}$ cm$^{-3}$ $\times 10^{-8}$ cm$^3$). Ketterle stressed that BEC, “Its most striking feature is a macroscopic occupation of the ground state of the system” [2]. The total number of atoms in a rarefied Bose gas is usually greater than $10^{11}$, but only $10^{-6}$ of them, can be trapped and cooled [1-6]. This fact shows that scientists still don’t know how to truly realize “the macroscopic occupation of the ground state” so far [1-6]. In addition, from Wikipedia, “quantum phase transition describes an abrupt change in the ground state of a many-body system. Such a quantum phase transition can be a second-order phase transition.” However, all previous articles failed to prove that the existing BEC experiments belong to second-order phase transition. These two facts reveal the shortcomings of the existing BEC research. Can we find a new way to get out of the present predicament? This article provides a new idea. Contrary to classical phase transitions, quantum phase transitions can only be accessed by varying a physical parameter—such as voltage. So the best way to achieve BEC is to measure capacitance at different voltages, which is a breakthrough discovery. When $V=370$ volts, the capacitance decreased from $C=2.2C_0$ to $C=C_0$ ($C_0$ is the vacuum capacitance), this result implies that almost all the Cs atoms (more than 98%) were to line up along the field, condensates contained up to $2.71 \times 10^{14}$ atoms (the total number of Cs atoms is $2.74 \times 10^{17}$). The waveform images on the oscilloscope clearly and truly show that large-scale BEC has been observed, this is the condensation in momentum space. The partition function formula (Eq.(1)), entropy formula (Eq.(2)), order parameter formula (Eq.(5)), condensate fraction formula (Eq.(9)) and atomic PDM formula (Eq.(11)) was first published in the history of physics, they constituted a complete theory of polar atoms. On the other hand, despite the relentless search for a non-zero PDM of an atom for more than 50 years, no conclusive results have been obtained so far [25-27]. Now an accurate and meaningful result has been obtained on the PDM of Cs atom. Experimental alkali material with purity 99.95% was supplied by Strem Chemicals Co., USA.

2. Our Innovation

A common view is that all kinds of atoms have no PDM because of their spherical symmetry, and there is no polar atom in nature except polar molecules. However, our research showed that alkali atoms form an exception. If alkali atom is a nonpolar atom, has this conclusion been verified by the experiment? Answer: no! Such experimental results have never been reported in the history of physics! In fact, this deep-rooted traditional concept is an unconfirmed assumption that misleads scientists all over the world. Although this assumption has never been suspected, it must be verified by accurate experiments. The title of the article is “Theoretical and experimental proof of alkali metal atoms as polar atoms”, which has been published in Journal of Material Sciences & Engineering (Impact Factor is 5.447) [7]. Reviewer comments pointed out that “The main goal of this submission is to prove the presence of polar atom in nature. This was done both theoretically and experimentally. The author has experimentally found that most alkaline atoms (Na, K, Rb and Cs) are polar atoms. He found this innovation by measuring their capacitance and temperature-dependence of their density. In fact, using Bose-Einstein condensation some decades ago, have shown that slim part of atom can be condensed ($10^7$) which is by far less than the normal conditions. The present author has modified Bose-Einstein condensation experiment and applied an electric field, which have a great impact on the production technique and on the performance in their monolithic type reactor system. The submission contributes to an increased knowledge either in theoretical or experimental parts. The subject of the above-mentioned submission is interesting and I suggest publishing the manuscript.” The fact shows that our innovative idea and experiments have been accepted by the scientific community.

The ways to achieve the macroscopic occupation of the ground state as follows. Our innovation lies in the application of an electric field, we don’t need a magnetic field and a laser. When the electric field was applied, alkali atoms become dipoles, they tend to align with the field because of the torque it experiences. The magnitude of the torque is $\tau = dE \sin \theta$, and the electric potential energy is $U = -dE \cos \theta$. When $\theta = 0$, the dipole is in equilibrium, and its potential energy is a minimum. But atomic collisions tend to disarrange the dipoles. From the entropy $S = 0$, critical voltage $V_c = 78$ volts. When $V < V_c$, many atoms are in random directions, this state has high entropy $S > 0$ and $C >> C_0$; when $V > V_c$, the atoms become aligned with the field, this state has low entropy $S < 0$, phase transition occurred. When $V >> V_c$, $C \approx C_0$, almost all the Cs atoms were to line up along the field, which is the macroscopic occupation of the ground state. These atoms have the same momentum, and this is condensation in momentum space.

This article is based on two previous articles, and this fact greatly enhanced the credibility of this work [8, 9]. In March 2016, when $V=390$ volts, the electric field produced condensates of $2.51 \times 10^{17}$ sodium atoms, and large-scale BEC has been observed [8]. The title of this article is “Bose-Einstein condensation in a vapor of sodium atoms in an electric field”. After a rigorous peer review, this article has been published in a professional journal (Physica B: condensed matter physics) [8]. Reviewer comments pointed out that “The author presented a good idea (using the critical
3. Theoretical Breakthrough

In theory, the 6s and 6p states of cesium atoms are not degenerate, so the expectation value of PDM is zero: 

\[ \langle \psi_{6s} | \sigma \cdot r | \psi_{6s} \rangle = 0 \]  

(\( d = \sigma \cdot r \) is the dipole moment operator) [13]. However, many physicists think that the quantum state \( \psi_{6s} \) doesn't describe an individual particle but an ensemble of particles with the same energy [13-15]. In the abstract of an article entitled "The statistical interpretation of quantum mechanics", this article has been published in Reviews of Modern Physics, L. E. Ballentine stressed that “the quantum state description to apply only to an ensemble of similar prepared systems, rather than supposing, as is often done, that it exhaustively represents an individual physical system. Most of the problems associated with the quantum theory of measurement are artifacts of the attempt to maintain the latter interpretation.” [14] D. J. Griffiths also stresses that “The expectation value is the average of repeated measurements on an ensemble of identically prepared systems, not the average of repeated measurements on one and the same system.” [16]

It convincingly proved that \( \langle \psi_{6s} | \sigma \cdot r | \psi_{6s} \rangle = 0 \) only means that the average PDM of large number of cesium atoms is zero, but doesn't mean that the PDM of individual Cs atom is zero.

The hydrogen atom is a typical example. “The shift in the energy levels of an atom in an electric field is known as the Stark effect.” [13] Normally the effect is quadratic in the field intensity, which corresponds to an induced electric dipole moment. L.D. Landau once stated that “the hydrogen atom forms an exception; here the Stark effect is linear in the field.” “The energy levels of the hydrogen atom, unlike those of other atoms, undergo a splitting proportional to the field (the linear Stark effect).” [17] Evidently, the linear effect corresponds to a PDM. This effect showed that the first excited state of hydrogen atom (n=2) has the PDM of magnitude 3e\(a_0\) = 1.59 \times 10^8 \text{e.cm} \( (a_0 \) is the Bohr radius) [17-19]. Landau once stated that "The presence of the linear effect means that, in the unperturbed state \( \psi_{2n} \), the hydrogen atom has a dipole moment whose mean value is 3e\(a_0\)" [17]. L.I. Schiff also stated that "It is also possible, as in the case of the hydrogen atom, that unperturbed degenerate states of opposite parities can give rise to a permanent electric dipole moment." [18] Griffiths emphasizes that "Spin is irrelevant to this problem, so ignore it, and neglect the fine structure." “Notice that the results are independent of the applied field, evidently hydrogen in its first excited state can carry a permanent electric dipole moment.” [16]

The four wave functions of the unperturbed state are \( \psi_{2n0}, \psi_{210}, \psi_{211} \) and \( \psi_{21-1} \). That is, \( d(\psi_{2n0}) = d(\psi_{210}) = d(\psi_{211}) = d(\psi_{21-1}) = 3e\(a_0\). However, although \( \psi_{21m} \) is four-fold degenerate, quantum mechanical calculations indicated that the expectation value of PDM is zero: \( \langle \psi_{21m} | \sigma \cdot r | \psi_{21m} \rangle = 0 \). Evidently, the zero result is inconsistent with above conclusion. Up to now, no quantum mechanical textbook explains this contradictory result. This fact shows that individual hydrogen (n=2) has a non-zero PDM, but quantum mechanics can't obtain this non-zero result in any way. It proved that \( \langle \psi_{21m} | \sigma \cdot r | \psi_{21m} \rangle > 0 \) only means that the average PDM of large number of hydrogen is zero, but an individual hydrogen (n=2) may have a non-zero PDM.

Recall that alkali atoms having only one valence electron in the outermost shell can be described as hydrogen-like atoms [19]. So \( \langle \psi_{6s} | \sigma \cdot r | \psi_{6s} \rangle = 0 \) only means that the average PDM of large number of alkali atoms is zero, but doesn't mean that PDM of individual alkali atom is zero. Alkali atoms (including cesium) may be polar atom doesn't conflict with quantum mechanics. Individual alkali atom is or is not polar atom must be determined by experiments.

The entropy and order parameter clearly described the order-disorder transition, which are the two most important concepts in BEC. The entropy of a system is a measure of the disorder of molecular or atomic motion. This is a thermodynamic quantity that is not easily measured. However, using a combination of theoretical and experimental techniques, entropy can in fact be accurately estimated. Consider a system composed of N cesium atoms, which are placed in an electric field E, \( \theta \) is the angle between d and E. Note that the collision between cesium atoms is always through their mass centers, and therefore the atomic nucleus has no contribution to the rotational energy of the atom. When orientation polarization occurs, its rotational energy can be neglected. The potential energy of the atom is \( e = -dE \cos \theta \). The orientation of cesium atoms can be changed continuously in the field. Note that \( \beta = 1/kT \) and the chemical potential \( \mu \approx 0 \) [6], the single-particle partition function is given by

\[
Z = \int_0^{2\pi} d\phi \int_0^\pi e^{iE\cos\theta} \sin \theta d\theta = \int_0^{2\pi} d\phi \int_0^\pi e^{iE\cos\theta} \sin \theta d\theta = 2\pi kT(e^{E/kT} - e^{-E/kT})/dE
\]

given by \( S = Nk(\ln Z + T \frac{\partial}{\partial T} \ln Z) \) [6]. Let the coefficient \( a = dE/kT = dV/kT \), we obtain
S (a) = Nk ln 2πe (e^a - e^{-a}) /a - a coth a (2)

The critical coefficient is a_0 = 2πe ≈ 17.08. Corresponds to a_0, the voltage V_c is called the critical voltage. When a_0 ≈ 2πe, e^{-a} = 0 and coth a_0 = 1, we obtain a simplified formula

S = Nk ln 2π e /a (3)

The entropy formula contains two fundamental constants in nature (π ≈ 3.14159 and e ≈ 2.71828), and there was ample reason to believe that it reflects the objective laws of BEC (see Figure 1).

![Figure 1](image)

Figure 1. S-V curve shows that BEC is a typical second-order phase transition, because the entropy doesn't show a discontinuity, this fact proved that it is also a quantum phase transition. Its inflection point is found: V_c = 8.7 volts.

Ehrenfest proposed a classification of phase transitions [6]. S – V curve shows that our BEC is a second-order phase transition, because the entropy doesn't show a discontinuity and its inflection point is found (Figure 1). The fact shows that the BEC experiment has no latent heat, it is continuous phase transition. Now we find the inflection point of S (a). From the second order derivative of Eq.(2) is zero, we obtain

S''(a) = coth^2 a + 2 coth a + 1/a^2 - 2 a coth a - 1 = 0 (4)

Note that coth a = (e^a + e^{-a}) / (e^a - e^{-a}), obviously, there is no analytical solution to this equation. From S''(1.9111860) = -6×10^{-8} < 0, S' (1.9111861) = 0 and S'' (1.9111862) = 2×10^8 > 0, we obtain its inflection point is a_0 = 1.9111861 ≈ 1.91. The voltage of the inflection point is V_c = 8.7 volts (for cesium) (see Figure 1).

Order parameter is a measure of the degree of order across the boundaries in a phase transition system; it normally ranges between zero in one phase and non-zero in the other phase. An example of an order parameter is the net magnetization in a ferromagnetic system undergoing a phase transition. For liquid/gas transitions, the order parameter is the difference of the densities. So the order parameter of BEC is

P = n k T \frac{\partial}{\partial E} \ln Z = n d L(a) (5)

where P is the polarization (the average dipole moment per unit volume), n is the density of alkali gas, d is the PDM of atom, L(a) = coth a - 1/a is Langevin function.

On the other hand, if a material is composed of polar molecules, the polarization also can be computed by the method of statistical mechanics. In order to avoid confusion with the differential symbol d, d_o represents the PDM of a molecule. In an electric field E = E_o, the potential energy of a polar molecule is U = -d_o E cosθ. In the heat equilibrium state, the number of molecules at θ per unit solid angle is proportional to e^{−U/kT}, the average dipole moment points in the direction of E_o < d_o > = d_0 < cosθ > E_o [20]. The average value of cosθ is

< cosθ > = \frac{1}{\pi} \int_0^\pi \cos d_o E \cos \theta /kT) \sin \theta d\theta = L(a), f = \frac{1}{\pi} \int_0^\pi \exp(d_o E \cos \theta /kT) \sin \theta d\theta (6)

where f is normalization constant. If n is the density of polar gas, the polarization is given by [20]

P = n d_0 < cosθ > = n d_0 L(a) (7)

Eq.(5) miraculously coincides with the result derived from the statistical mechanics of electromagnetics, which shows that our theory is self-consistent [20]. The creation of ordinal parameter formula is a milestone in the history of BEC, because this formula has not been deduced by any previous BEC article or work [1-12]. < cosθ > = L(a) shows that L(a) equals the percentage of alkali atoms lined up along the electric field, it equals the condensing fraction of BEC.

How to separate polar and non-polar substances (molecules or atoms) experimentally?

Variation of capacitance with temperature offers a means of separating polar and nonpolar atom experimentally, but no one has done this discriminating experiment until now. If alkali atom is non-polar atom, its capacitance should be independent of temperature, because the nucleus is located at the center of the electron cloud. Conversely, if alkali atom is a polar atom, its capacitance is related to temperature. The electric susceptibility is defined as χ_e = C/C_0-1. For polar atoms or molecules, the electric susceptibility is given by χ_e = nα + n d L(a)/ε_0 E, where α is the atomic or molecular polarization. When α << 1, L(a) ≈ a/3, you can get a familiar formula. [21]

For polar atom χ_e = nα + n d^2 /3k ε_0 T, for nonpolar atom χ_e = nα (8)

4. Experiments and Interpretation

Experiment 1: accurate measurement of the density of cesium gas.

Chemists think that the measurement of the density of gases is a difficult problem. In the textbook of physical chemistry experiment, the experiment of measuring molecular dipole moment is introduced in detail, and this difficulty in measuring the density of gas is particularly stressed. Two coaxial cylinders constitute a traditional cylindrical capacitor. However, because of the irregular orientation of alkali atoms (including cesium) in the electric field, it is impossible to use this capacitor to carry out the BEC experiment (see Figure 3).
In order to achieve BEC, an ingenious apparatus has been adopted. Our experimental apparatus is a closed glass container resembling a Dewar flask (Figure 2). The internal and external diameters of the container are D_a=5.70cm and D_b=8.22cm. The external and internal surfaces are plated with silver, showed by a and b respectively (see Figure 2). The length of the silver layer is L=28.2cm. As a more convenient method, replacing plate silver, we can use aluminum foil wrapped the internal and external surface of the glass container, and the two aluminum cylinders constitute a cylindrical capacitor. Note that unlike the traditional cylindrical glass capacitor, the two aluminum cylinders must be posted on the glass. This capacitor is connected in series by two capacitors. One is called C′ and contains the Cs gas of thickness H_0=9.6 mm; the other is called C” and contains the glass medium of thickness h=1.5mm. The total capacitance is C = C′C”/(C′+C”), where C” and C can be directly measured.

When the capacitor was empty, it was pumped to vacuum pressure P ≤10^{-9} torr for 20 hours. The aim of the operation is to remove impurities such as oxygen adsorbed on the inner walls of the container. The vacuum capacitance is C_0 = (56.6±0.1)pF. Next, a small amount of Cs material (about 10 grams) was put into the container, and it is again pumped to P ≤10^{-9} torr. A cylindrical capacitor filled with Cs gas and surplus liquid cesium has been obtained (Figure 2). Measuring instrument is a digital capacitance meter (DM6031A made in Shenzhen, China). The measuring voltage is V_0=1.2 volts, the measuring frequency was 800Hz, and the definition was 0.1 pF (C≤200pF), 1pF (200pF≤ C ≤ 2nF) or 10pF (2nF≤ C ≤20nF). The capacitor is put into a temperature-control stove, and keep it at T_0 =473K for 6 hours. Its capacitance becomes C_{i0} = (5290±10) pF. The saturated vapor pressure of Cs atoms is P = 10^{6.949-3833.7/T} psi (473K≤T ≤ 623K), where 1 psi=6894.8 Pa [22]. Atomic polarizability of mercury is α = (9/2) 4πR^3/k ε_0, where R =7.37×10^{-3} cm, ε_0=8.85×10^{-12} C^2/Nm^2, and k=1.99×10^{-3} cm/N. Therefore, the ideal gas law, the density of cesium gas equals n_0 = P/kT_0 = 7.37×10^{-13} cm^{-3}. The statistical error is (∆n/n)^2 = (AP/P)^2 + (∆T/T)^2, due to ∆T = 0.5K and ∆P/P≤ 0.02, so ∆n/n ≤ 0.03. Systematic error is ∆n/n = 0.03, the density of cesium gas is n_0 = [7.37×0.22(stat) ±0.22 (syst)] ×10^{16} cm^{-3}.

Experiment 2: accurate measurement of the capacitances of Cs and Hg gas at different density.

A cylindrical capacitor fills with Cs gas and surplus liquid cesium, another capacitor fills with Hg gas and surplus liquid mercury (see Figure 2). Their vacuum capacitance is C_{01} = 56 pF (for Cs gas) and C_{01} = 51.8 pF(for Hg gas) respectively. After fill with Cs or Hg materials, the capacitances become C_{01} = 109.8pF (for Cs gas) and C_{01} = 52.1pF (for Hg gas). We measure the capacitances and density at different temperatures. The experiment showed that the capacitances of Cs gas increased sharply with the increase of temperature, but the capacitance of Hg gas, C_{01} =52.1pF, remains constant, and χ_e =0.006. Table 1 gives a complete experimental data of cesium. The experimental results are shown in Figure 4.

Atomic polarizability of cesium is α/4π = 59.6 ×10^{-30}m^3, and α=74.9×10^{-28}m^3 [20]. If cesium atom is non-polar atom (d=0), when T=473K, and n_0=7.37×10^{22} m^{-3}, the susceptibility of Cs atoms is only χ_e = n_0 α = 5.52×10^{-3}, the capacitance of Cs vapor is only C_e = 56.003pF. However, the experiment shows that when T=473K, χ_e =93.4, and C_e =5290pF. Obviously, the atomic polarizability and capacitance of Cs vapor are inconsistent with the experimental results! This experiment proved that cesium atom is a polar atom.

Table 2 gives a complete experimental data of mercury gas. The formula of saturated vapor pressure of mercury atoms is P=10^{7.752-3063.9/T} mmHg (673K≤T≤1573K). Because the temperature (673K) exceeds the furnace temperature, we use the experimental values of mercury saturated vapor pressure [22]. Atomic polarizability of mercury is α = (9/2) 4πR^3 =
19.8×10^{-20} m^3, where R = 1.52×10^{-10} m is the radius of Hg atom, 9/2 is the correction of quantum mechanics[20].

When T=497K, and the density of Hg vapor up to n=6.819×10^{10} m^{-3}, but the theoretical value of the susceptibility is only \( \chi_e = na = 1.36×10^{-4} \), the capacitance of Hg vapor is only C= 51.81pF= C_{01}, remains constant (see Table 2).

| P (Pa)     | n(10^{12} m^{-3}) | C (pF) | \( \chi_e \) |
|------------|-------------------|--------|--------------|
| 483        | 562.8             | 762.9  | 1026         | 1184 |
| 477        | 656.4             |        |              |
| 481        | 762.9             | 886.1  | 1026         | 1184 |
| 485        | 1.139             | 1.312  | 1.507        | 1.725|
| 489        | 5290              | 5560   | 6380         | 6930 |
| 493        | 7880              | 8600   |              |
| 497        | 8960              |        |              |

Thus a capacitor filled with Cs gas at a fixed density is obtained (i.e. ensure the number of Cs atoms remained constant). We adopted an ingenious technique (see Figure 5).

The two tables formed a sharp contrast because Cs atom is polar but Hg atom is non-polar.

\[ \text{Experiment 3: accurate measure capacitance of cesium gas at various voltages.} \]

The best way to realize BEC is to measure the capacitance at different voltages. The experimental instrument is a cylindrical glass capacitor filled with cesium gas at a fixed density \( n_2 \), and \( n_2 < n_0 \). The experimental temperature is maintained at 343K. In order to ensure that the capacitor is filled with Cs gas at a fixed density (i.e. ensure the number of Cs atoms remained constant), we adopted an ingenious technique (see Figure 5).

The capacitance is measured by the comparison method, \( C_{20} \) is the capacitor to be measured, its vacuum capacitance is \( C_{20} = (68.0±0.1) \text{pF} \), and \( H_{2} = 7.2 \text{mm} \). \( C_{01} = 1878 \text{pF} \) is a standard capacitor, \( C_{2} \) and \( C_{4} \) are series. Two signals \( V_c(t) = V_{co} \text{cos} \) and \( V_c(t) = V_{co} \text{cos} \) were measured by a two channel digital oscilloscope (Tektronix TDS 210 USA). The signal \( V_c(t) \) is applied to the capacitor \( C_2 \), the signal \( V_c(t) \) is applied at both ends of \( C_2 \) and \( C_4 \). Because \( C_2 \) and \( C_4 \) series, so \( (V_c-V_c) = C_2/C_4 \) and \( C_2 = (V_{co}/V_{co} - 1)C_4 \). The voltages \( V_{co} \) could be adjusted from zero to 800 V. The measurement was started in 0.01 V. When \( V_c \leq 0.4 \text{ V} \), the capacitance \( C_{21} = 150 \text{pF} \) is approximately constant. When \( V_c \gg V_c \), the peak difference of the two signals is large, this image means that \( C_2 >> C_{20} \), this state is far from BEC. For example, when \( V_c = 7.08 \text{V} \), \( V_{co} = 6.6 \text{V} \), \( C = (7.08/6.6 - 1)C_4 = 136.5 \text{pF} >> C_{20} \). If almost all the dipoles in a gas were to line up with an external electric field, this effect is called the saturation polarization. The \( C - V \) curve shows that the saturation polarization of cesium gas is easily observed when \( V_c >> V_c \) (Figure 6). When \( V_{co} = 384\text{V} \), \( V_c = 71.0 \text{pF} \), \( C = 71.0 \text{pF} = C_{20} \). The oscilloscope suddenly shows that the two signals have the same phase, and they almost overlap. This waveform images clearly and truly show that large-scale BEC has been observed.

\[ \text{Figure 5. This diagram shows the perfusion method. Let the left capacitor to connect a vacuum pump through a glass tube. In this glass tube, a small magnetic hammer and a sealed cesium sample are put together. When the pump works, the magnetic hammer is raised by a magnet outside the tube. The magnetic hammer is suddenly release, which breaks the bottle sealed cesium sample, and then the tube is sealed at B point. Next step, the two containers are put into a heating furnace, and keep them at a certain temperature for 6 hours. After Cs gas is filled with the capacitor, glass tube is sealed again at A point. Thus a capacitor filled with Cs gas at a fixed density is obtained (i.e. ensure the number of Cs atoms remained constant).} \]

\[ \text{Figure 6. Experimental C – V curve shows that when } V_c \leq 0.4 \text{ volts, } C_{21} = 150 \text{pF} >> C_0 \text{ when } V_c = 370 \text{volts, } C_{21} = 71.0 \text{pF} = C_{20}. \text{ This result implies that almost all the Cs atoms (more than 98.7%) were to line up along the electric field, condensates kept up to 2.71×10^{13} \text{ atoms, BEC becomes visible and tangible.} } \]

\[ \text{Previous papers usually provide color images of the velocity distribution of trapped atoms, but they are simulated by computer. Many scientists think that these images are “false color images”. [23] Unlike them, the waveform images on the oscilloscope show that we created BEC with 2.71×10^{13} \text{ atoms, BEC becomes visible and tangible.} } \]

\[ \text{Condensate fraction: } L(a) = a(C - C_0)/n \text{ or } C = \eta L(a)a + C_0 \]

where \( \eta = S_{nd^2}/k\text{TH} \) is the capacitance constant. When \( V_c \leq 0.4\text{V} \), \( a < 0.1 \), \( L(a) = a/3 \) and \( \eta = 3(C_{21} - C_{20}) = 246 \text{pF} \). When \( V_c = 370\text{V} \), \( C_{21} - C_{20} = 3.0 \text{pF} \), \( L(a) = 1-1/a_2 \), \( a_2 = \eta L(a)/(C_{21} - C_{20}) \).
Why our classical theory can provide satisfactory explanation for the condensation of quantum gas?

R.P Feynman gave a wonderful demonstration. First, the photon as an example, he once stated that photons are analogous to the Schrödinger equation for the electron, the photon equation is just the same as Maxwell’s equations for the electromagnetic field, and the wave function is the same as the vector potential A [24]. Second, in order to explain superconductivity, he remarked that a Cooper pair acts as a boson, many bosons in the same state, “ψψ∗ can be interpreted as the electric charge density”, and “J (the current of probability) becomes directly the electric current density”, so the electromagnetic equation can be used instead of Schrödinger equation to explain superconductivity [24]. Similarly, alkali atoms (including cesium) in a rarefied gas are non-interacting bosons, when \( V >> V_{c} \) almost all bosons (more than 97%) are in exactly the same state. Therefore, according to Feynman, “the quantum physics is the same thing as the classical physics” [24], so our classical theory provided a satisfactory explanation. Attention please, the necessary condition is many alkali atoms are in the same state, an applied electric field can satisfy this requirement. Despite scientists have taken many advanced technologies, however, they can't put the alkali atoms in the same state, so they missed this significant discovery.

5. Discussion

5.1. The Theoretical Evidence That Cesium Atoms Do Not Violate Time Reversal and Parity Symmetry

Experiments to search for PDM of atoms began half a century ago. In all experiments, they measured the spin resonance frequency \( \nu \) of individual atom by \( h \nu = 2\mu B = 2\mu E \), where \( h \) is Planck’s constant, \( \mu \) and \( d \) is magnetic and electric dipole moments [25-27]. But “experimental searches for PDM have so far yielded null results.” [25] This fact shows that this formula is not successful, because it measured the microscopic quantity \( d \) by using another microscopic quantity

\[ n = (C_2 - C_{20}) V_2 \frac{L(a)}{n} \]

\[ \text{Table 3. Bose-Einstein condensation of cesium atoms at different voltages (V)} \quad T=343K. \]

| (volt) | (C PF) | \( a \) | L(a) | P(10^4 e.cm^2) | N_c(10^15) | S (+N_k) |
|-------|--------|-------|------|----------------|-------------|----------|
| 0.01  | 150    | 2.2x10^-3 | 7.3x10^-4 | 9.3x10^-4 | 0.002 | 2.531 |
| 0.40  | 150    | 0.087  | 0.029 | 0.037 | 0.0081 | 2.530 |
| 8.7   | 135.1  | 1.91   | 0.5213 | 0.6625 | 1.431 | 2.079 |
| 78    | 81.6   | 17.08  | 0.9414 | 1.196 | 2.584 | 0 |
| 220   | 73.0   | 48.2   | 0.9776 | 1.242 | 2.683 | 0 |
| 276.4 | 72.0   | 60.5   | 0.9835 | 1.250 | 2.707 | -1.037 |
| 370   | 71.0   | 81     | 0.9876 | 1.255 | 2.711 | -1.265 |
| ∞     | 68.0   | ∞      | 1.0    | 1.271 | 2.745 | -1.557 |

\( n_0 = 7.37 \times 10^{22} m^{-3} \). Using BEC data, we obtain the same result:

\[ d_{Cs} = 2.95 \times 10^{-29} \text{C.m} = 1.84 \times 10^{-6} \text{e.cm}, \]

where \( (C_{22} - C_{20}) = 3 \text{pF}, V_2 = 370 \text{V}, L(a) = 0.9876, S_{e} = 5.53 \times 10^{-4} \text{m}^2 \) and \( n_0 = 6.89 \times 10^{20} \text{m}^{-3} \). Since the density \( n_2 \) is unknown, from Eq.(9) the density of alkali gas under arbitrary vapor pressure is

\[ n = (C_2 - C_{20}) V_2 L(a) \frac{n_0}{n} \frac{L(a)}{S} \]

\[ \text{The density is } n_2 = 6.89 \times 10^{14} \text{cm}^{-3}, \text{where } a = a_0 V_0 T_0 S_{0} \text{e.cm} \text{ and } L(a) = 0.0476. \text{Considering all error, } n_2 = [6.89 \pm 0.41(\text{stat}) \pm 0.34(\text{sys})] \times 10^{14} \text{cm}^{-3}. \text{The trapping volume reached } S_{H} = 398.2 \text{cm}^2, \text{the total number of Cs atoms is } N_{\text{Cs}} = n_{2} S_{H} = 2.745 \times 10^{17}, \text{and the number of condensate atoms is } N_{\text{Cs}} = N L(a). \text{From the formula (9), atomic PDM can be deduced} \]

\[ d = (C - C_0) V L(a) n S \]

\[ \text{Table 3. Bose-Einstein condensation of cesium atoms at different voltages (V)} \quad T=343K. \]

\[ N_{\text{Cs}} \] The number of condensed atoms, \( S \) — The entropy of Cs gas, \( P \) — The order parameter of BEC. In fact, ultralow temperature is in order to make Bose gas phase transition, this article used the critical voltage to achieve the phase transition, and therefore the ultralow temperature is not necessary for observation of BEC. By mere guesswork, scientists generally believed that alkali atoms don't have PDM, they are nonpolar atom. In fact, this is an idealist philosophy. Our experiment shows that the susceptibility of alkali atoms is \( n = n_0 + n d^2/3kT \), so alkali atom is polar atom. This research proved that the condensate fraction of BEC is \( L(a) = (C - C_0)/2S_0 n S \text{L(a)ns}, \text{where } C - C_0 \text{ is the change of the capacitance. If } (C - C_0) \neq 0, \text{the experiment proves the PDM } d \neq 0, \text{condensate fraction } L(a) \neq 0, \text{so BEC occurs} \); conversely, if \( (C - C_0) = 0 \), we obtain \( d = 0 \) and \( L(a) = 0 \), BEC doesn't occur. Since scientists never measured the capacitance of alkali atoms, so they missed this significant discovery.
v. In the case of cesium, famous physicist Weisskopf reports a null result of early experiments: $d_{Cs} < 3.7 \times 10^{-20}$ cm (1968). [27] A representative result as follows: $d_{Cs} = (-1.8 \pm 6.7 \text{ (stat)} \pm 1.8 \text{ (syst)}) \times 10^{-20}$ cm [26]. Optical pumping is used to spin polarize the Cs ground state. Evidently, this method is difficult to measure the PDM of Cs atom because its error is 3.7 times as large as the measured value [26]. However, measuring the average kinetic energy of a gas molecule using the temperature is easy: $E = 3kT/2$. Similarly, the measurement of the PDM of an atom using the change of the capacitance is easy: $d = (C - C_0)/V/L(a) \times nS$

According to quantum theory, an atom in its ground state at most has an extremely small PDM, $d \leq e \times 10^{-20}$ cm, it points along the nuclear spin axis and arise mainly from the nuclear spin. Under time reversal the direction of the spin changes, while the direction of the PDM does not change. Therefore, the PDM of an atom would violate time reversal symmetry. By the CPT theorem it also implies a violation of CP symmetry [25]. In short, if the PDM of atom violate time reversal symmetry, it must have two characteristics: the first is very small, $d = \leq e \times 10^{-20}$ e. cm, and the second is arises from the nuclear spin.

However, the linear Stark effect of hydrogen atom provides a new example. This effect showed that the hydrogen atom (n=2) has a PDM of magnitude $3e_{\lambda_0} = 1.59 \times 10^{-3}$ e.cm. Since the nuclear spin was completely irrelevant to the calculation of the PDM, and the nuclear radius ($a_0 = 0.53 \times 10^{-8}$ cm) is far greater than the nuclear radius ($t_0 = 10^{-11}$ cm), so this PDM has nothing to do with the nuclear spin, and only arises from the asymmetrical charge distribution of hydrogen atom. L.D. Landau once stated that the hydrogen atom has a dipole moment whose mean value is $3e_{\lambda_0}$. “This is in accordance with the fact, in a state determined by parabolic quantum numbers, the distribution of the charges in the atom is not symmetrical about the plane z=0” [17]. So similar to the first excited state of hydrogen atom, the PDM of Cs atom doesn't arise from the nuclear spin but from asymmetrical charge distribution, and it doesn't violate time reversal and parity symmetry. [25-27]

5.2. Theoretical Evidence That the Linear Stark Effect of Cesium Atoms Cannot Be Observed

This is a challenging question. As two concrete examples, first let us deal with the fine structure and the linear Stark shifts of the hydrogen (n=2). The wavenumber of the fine structure of the hydrogen (n=2) is $\Delta W = 1.59 \times 10^{-8}$ e.cm. When $E = 10^2$ V/cm, $\Delta W = 1.59 \times 10^{-8}$ eV, this corresponds to a wavenumber of $12.8 \times 10^{-8}$ cm$^{-1}$. So the linear Stark shifts is $\Delta \lambda = \Delta W \lambda / \hbar c = 12.8 \times (656.3 \times 10^{-7})^2 = 0.55$ nm, and the linear Stark shift of the hydrogen (n=2) is easily observed [28]. However, when $V = 370V$, the most field intensity is $E_{max} = 514V/cm$. When the external electric field increases to $E = E_{max}$, almost all the Cs atoms (more than 98.7%) were to line up along the field, Cs vapor no longer absorb energy, this Stark effect will not occur. if the PDM of cesium is $d_{Cs} = 1.84 \times 10^{-8}$ e.cm, and the most splitting of the energy levels is $\Delta W_{max} = d_{Cs}E_{max} = 9.46 \times 10^{-10}$ eV. This corresponds to a wavenumber of $\Delta W_{max} / \hbar c \approx 0.076$ cm$^{-1}$. On the other hand, observed values for a line pair of the first primary series of Cs atom (Z=55, n=6) are $\lambda_1 = 894.3$ nm and $\lambda_2 = 852.1$ nm [28]. So the most linear Stark shift of Cs atoms is only $\Delta \lambda = \Delta W (\lambda_1 + \lambda_2) / 4 \hbar c = 0.0057$ nm. It is so small, in fact, that a direct observation of the linear Stark shifts of Cs atom is not possible by conventional spectroscopy!

5.3. The True Meaning of Einstein's Prophecy

In 1924, S. N. Bose, a physicist from India, wrote to Einstein describing his work on the statistical mechanics of photons. Einstein followed up Bose’s work by generalizing it to non-relativistic particles with non-zero mass, and in 1925 he predicted the phenomenon now known as Bose-Einstein condensation. The term ‘condensation’ often implies a condensation in configuration space. However, for Bose Einstein condensation it is a condensation in the momentum space, because the result of this condensation doesn't produce crystals. In 1925 Einstein pointed out that “I maintain that, in this case, a number of molecules steadily growing with increasing density goes over in the first quantum state (which has zero kinetic energy) while the remaining molecules distribute themselves according to the parameter value $\lambda=1$.” “A separation is effected; one part condenses, the rest remains a saturated ideal gas’ ($A=0$, $\lambda=1$).” [29] The famous physicist Abraham Pais emphasized that “this is now called Bose-Einstein condensation.” [29] If $1/10$ is used as “one part” of this gas, now estimate how many atoms are contained in this part of the Bose gas. In order to achieve BEC, the suitable density of Bose gas is $10^{13} - 10^{15}$ cm$^{-3}$, and the volume of Bose gas is usually greater than $10$ cm$^3$, and therefore the total number of atoms in the Bose gas is usually as high as $10^{14}$ or more. So according to Einstein's prediction, the number of condensed atom should be at least greater than $10^{13}$. But in the past, the number of condensed atoms is still very small (<10$^4$). This fact proves that although physicists have the ability to do BEC experiments, they still need to be more accurate understanding of the true meaning of Einstein's prophecy.

F. Wilczek, he was awarded the Nobel Prize in 2004, once stated that “The primary goal of fundamental physics is to discover profound concepts that illuminate our understanding of nature. Discovering new particles, as such, is secondary.” [30] Probably, “polar atom” is such a concept.

6. Conclusion

6.1. The Meaning of the Expected Value of Atomic PDM

In theory, $\langle \psi | \hat{e} | \psi \rangle = 0$ only means that the average PDM of large number of alkali atoms is zero, but doesn't mean that the PDM of individual alkali atom is zero. Despite 6s and 6p states of cesium are not degenerate, but Cs may be polar atom doesn't conflict with quantum mechanics because it is hydrogen-like atom.
6.2. All Kinds of Alkali Atoms Are Polar Atom

All kinds of alkali atoms are non-polar atom, which is an untested hypothesis, and it must be verified by accurate experiments. The capacitance at different temperatures and density are measured, our experiment proved that Cs atom is polar atom, because its electric susceptibility is given by $\chi = n a + n d l(a)/\varepsilon_0 E$. But in the past, scientists have never measured the capacitance of alkali atoms, so they missed this significant discovery.

6.3. Three Main Features of BEC

BEC has three main features: BEC is a macroscopic occupation of the ground state of the system; BEC is a condensation in momentum space; Bose gas would undergo a phase transition. Due to our experiments are fully in line with these three main features, although laser cooling techniques are not used, they are an ideal BEC. Ultra-low temperature is in order to make Bose gas phase transition, and we used the critical voltage to describe phase transition, and therefore ultra-low temperature is not necessary for observation of BEC.

6.4. The Entropy Is the Most Important Concept in BEC

The entropy of a system is a measure of the disorder of molecular or atomic motion. No doubt, it is the most important concept in BEC. This formula, $S = N k ln (2\pi e/a)^2$, contains two fundamental constants in nature ($\pi = 3.14159$ and $e = 2.71828$), and there was ample reason to believe that it reflects the objective laws of BEC.

6.5. Polar Atom will Be Candidate for Quantum Computers

The BEC is a typical second-order phase transition because entropy doesn't show discontinuity, according to Wikipedia, this BEC experiment is also equivalent to a quantum phase transition. Now scientists generally assume that polar molecules may be used as candidate materials for quantum computers. In the future, polar atoms will replace polar molecules as candidate materials for quantum computers because they have a smaller moment of inertia.

6.6. Theoretical Evidence That Our Theory Is Self-consistent

Order parameter is a measure of the degree of order across the boundaries in a phase transition system. Order parameter of BEC is the polarization. Entropy and order parameter clearly described order-disorder transition. This formula miraculously coincides with the result derived from the statistical mechanics of electromagnetics, which shows that our theory is self-consistent.

6.7. Our Classical Theory Provided Satisfactory Explanation

Alkali atoms (including cesium) in a rarefied gas are non-interacting bosons, when $V \gg V_c$, almost all alkali atoms are in exactly the same state. Therefore, according to Feynman, “the quantum physics is the same thing as the classical physics”, so our classical theory provided a satisfactory explanation for BEC experiment.

6.8. Cs atom Has Non-zero PDM

The PDM of Cs atom has been accurately measured: $d_{Ce} = [1.84 \pm 0.15 \text{(stat)} \pm 0.11 \text{(syst)}] \times 10^{-8} \text{ e.cm}$. The PDM doesn't arise from the nuclear spin but arises from asymmetrical charge distribution, and it doesn't violate both time reversal and parity symmetry.

6.9. We Created Five Innovative Formulas

Partition function formula, entropy formula, order parameter formula, condensate fraction formula and atomic dipole moment formula not only proves that our experiment is an ideal BEC, but also shows that the complete theory of polar atoms has been obtained. The five innovative formulas were first reported in the history of physics.

6.10. BEC will Become Enjoyable and Fascinating

Our experiment is easily repeated in other laboratories, because the details have been described in this article. Once scientists completed these measurements, they will obtain the same results as our experiments. They will discover that a large-scale BEC of cesium atoms is easily observed when an electric field is applied! The publication of the article will lead to a revolution in BEC research and condensed matter physics, marking mankind will enter a new era of polar atoms. We fervently hope that this article will make BEC research to become enjoyable and fascinating. In the face of the energy crisis and the potential threat of climate change, we want to use this article to express our enthusiasm for the readers.

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