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Kabir, Mukul et al. “Effects of vacancy-solute clusters on diffusivity in metastable Fe-C alloys.” Physical Review B 82.13 (2010): 134112. © 2010 The American Physical Society.

http://dx.doi.org/10.1103/PhysRevB.82.134112

American Physical Society

Final published version

http://hdl.handle.net/1721.1/60898

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Effects of vacancy-solute clusters on diffusivity in metastable Fe-C alloys

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(Received 19 April 2010; revised manuscript received 24 August 2010; published 15 October 2010)

Diffusivity in defected crystals depends strongly on the interactions among vacancies and interstitials. Here we present atomistic analyses of point-defect cluster (PDC) concentrations and their kinetic barriers to diffusion in ferritic or body-centered-cubic (bcc) iron supersaturated with carbon. Among all possible point-defect species, only monovacancies, divacancies, and the PDC containing one vacancy and two carbon atoms are found to be statistically abundant. We find that the migration barriers of these vacancy-carbon PDCs are sufficiently high compared to that of monovacancies and divacancies. This leads to decreased self-diffusivity in bcc Fe with increasing carbon content for any given vacancy concentration, which becomes negligible when the local interstitial carbon concentration approaches twice that of free vacancies. These results contrast with trends observed in fcc Fe and provide a plausible explanation for the experimentally observed carbon dependence of volume diffusion-mediated creep in ferritic (bcc) Fe-C alloys. Moreover, this approach represents a general framework to predict self-diffusivity in alloys comprising a spectrum of point-defect clusters based on an energy-landscape survey of local energy minima (formation energies governing concentrations) and saddle points (activation barriers governing mobility).

DOI: 10.1103/PhysRevB.82.134112

I. INTRODUCTION

In crystalline solids, diffusion proceeds as atoms migrate via lattice defects that may be thermally intrinsic or processing induced.1 There are many diffusion mechanisms, of which vacancy and interstitial diffusion are the most common.2 In defected crystals and crystalline alloys, prediction of macroscale diffusivity via microscopic mechanisms is further complicated by the potential formation of clusters comprising multiple types and species of point defects (i.e., point-defect clusters, or PDCs). Typically, the concentration of these PDCs is not known or experimentally accessible, even when the overall concentration of atomic species and lattice vacancies is reasonably well quantified. Thus, quantification of how PDCs impact thermally activated processes, such as diffusion-mediated mechanical failure, presents a scientific challenge of technological relevance. Here, we present a computational framework to predict the self-diffusivity in interstitial alloys comprising appreciable PDC populations. We focus our models and simulations on a specific alloy, body-centered-cubic (bcc) iron that approaches supersaturation of both carbon and vacancies.

The strengthening effects of interstitial carbon solute atoms in (i.e., ferritic or bcc) Fe-C alloys are well understood, owing chiefly to the interaction of C with crystalline defects (e.g., dislocations3 and grain boundaries3,4) to resist plastic deformation via dislocation glide. High-strength steels developed for current energy and infrastructure applications include alloys wherein the bcc Fe matrix is thermodynamically supersaturated in carbon. This degree of supersaturation ranges from 0.3 at. % C upon annealing (670 K) (Ref. 4) to 5 at. % C in volumes that include ferritic grain boundaries.3 It has been demonstrated that the strong affinity of C to Fe vacancies also maintains an order-of-magnitude supersaturation of vacancies via the formation of PDCs comprising multiple C atoms and multiple Fe vacancies.5,6 This vacancy supersaturation is also stabilized by other solute atoms including hydrogen.7,8 Self-diffusion in bcc Fe is governed chiefly by vacancy migration, as the alternative self-interstitial migration mechanism is suppressed in (nonirradiated) bcc Fe due to its significantly higher formation energy.6 Thus, although it is not yet experimentally feasible to determine the spatial or temporal distribution of these PDCs in such defected crystals, it is reasonable to anticipate that processes requiring self-diffusion in the Fe lattice would be affected by the presence and mobility of these PDCs.

Here we report the effect of solute (carbon) concentration on self-diffusion in a dislocation-free volume of single crystal bcc Fe. The formation and binding energies of statistically abundant PDCs are calculated via electronic [first-principles density functional theory (DFT) (Refs. 9 and 10)] and atomistic (many-body semiempirical potential developed previously for this system11,12) approaches. Activation barriers of migration are calculated via the nudged elastic band (NEB) method.13,14 Such calculations are nontrivial, given the many possible and nonredundant configurations and diffusive pathways available to PDCs of increasing size. Self-diffusivity is then calculated for a system comprising statistically abundant PDCs, as a function of temperature and overall alloy composition. We find that in bcc Fe supersaturated with both carbon and vacancies, three PDCs are statistically abundant: monovacancies (Va1), divacancies (Va2), and Va1C2 clusters. As the migration barrier of the Va1C2 cluster significantly exceeds that of the carbon-free PDCs, the self-diffusivity in such defected volumes is controlled by the migration of Va1 and Va2. Thus, self-diffusivity decreases with increasing at. % C and becomes negligible when the carbon concentration exceeds twice that of vacancies. The present results are consistent with the experimental observa-

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tation that macroscopic creep strain rate decreases with increasing carbon content in bcc Fe-C alloys.\textsuperscript{15} Moreover, this framework provides a general approach to relate microscopic point-defect interaction mechanisms to macroscopic properties in crystals of increasingly realistic defect composition, based on an energy-landscape perspective of calculated energy minima and saddle points.

II. METHODOLOGY

Our calculation of migration energies combines three elements of calculation. First, PDC energetics are calculated from DFT and a modified interatomic potential (constructed from such DFT calculations).\textsuperscript{2} Second, PDC concentrations are determined from a statistical model, to identify statistically abundant point-defect clusters as a function of overall composition. Third, PDC migration mechanisms and barriers are ascertained from NEB transition state pathway sampling.

Formation energies, binding energies, and migration barriers of PDCs were calculated via both DFT (Refs. 9 and 10) and modified Finnis-Sinclair (MFS) potential.\textsuperscript{11} Formation energy of PDCs comprising \( x \) Fe vacancies (Va) and \( y \) octahedral carbon (C) interstitials is given by\textsuperscript{6,16}

\[
E_{\text{Va,C}}^f = E(Va,C_y) - E_0 - xE(Fe) - yE(C),
\]

where \( E(Va,C_y) \) and \( E_0 \) are the free energies of the supercells (neglecting vibrational entropy) with and without the PDC present, respectively. \( E(Fe) \) is the reference energy of bulk, defect-free bcc Fe and \( E(C) \) is the reference energy of a C atom in vacuum. The corresponding binding energy of the PDC cluster, i.e., the energy gain in forming the PDC from the respective individual point defects is calculated as\textsuperscript{16,17}

\[
E^b(Va,C_y) = xe^{E_{\text{Va}}} + ye^{E_{\text{C}}} - E_{\text{Va,C}}^f,
\]

where \( E^b \) represents the formation energy of C at the bcc octahedral interstitial site. The spin-polarized DFT calculations were conducted using the VASP code\textsuperscript{18,19} with the Perdew-Burke-Ernzerhof exchange-correlation functional\textsuperscript{20} and projector augmented wave pseudopotential\textsuperscript{21} at an energy cutoff of 400 eV. The calculations were performed with 128-atom supercells and geometry optimization terminated when the force on each atom was <5 meV/Å. Details of the MFS potential are given elsewhere.\textsuperscript{11}

Calculation of the PDC spectrum requires a common lattice for the many types of defect clusters, as we have described previously.\textsuperscript{6,16} Similarly, for the present case the defect concentration \( C_i \) of a particular PDC (Va,C_y) is defined with respect to the bcc Fe lattice

\[
C_i = (C_v)^y(C_f)^x \left[ \frac{\alpha_i}{\exp\left(-\frac{E_i^b}{k_BT}\right) + 1} \right],
\]

where \( \alpha_i \) is defined such that \( \alpha_iN \) (\( N \) being the total number of bcc lattice positions) represents the number of indistinguishable configurations in the lattice for that particular defect cluster species.\textsuperscript{16}

We constrained the total carbon \( \nu^c_{\text{total}} \) and vacancy concentration \( \nu^v_{\text{total}} \) to be equal to the sum over all PDC species,

\[
\nu^c_{\text{total}} = \sum_{\text{clusters}} yC_{\text{Va},C_y},
\]

\[
\nu^v_{\text{total}} = \sum_{\text{clusters}} xC_{\text{Va},C_y}.
\]

This approach enables calculation of PDC concentrations for any arbitrary, total concentration of carbon and vacancies.

Under the transition state approximation, the diffusion rate of a defect species in solids can be written as

\[
\Gamma = \tilde{\rho}e^{-\frac{E_m}{k_BT}},
\]

where \( E_m \) is the migration barrier of the process (i.e., the energy required to carry the defect from an initial equilibrium position to a saddle point) and \( \tilde{\rho} \) is the frequency prefactor. For temperatures far below the melting temperature, the atomic positions in crystalline solids do not deviate greatly from their equilibrium positions. Hence it is reasonable to invoke the harmonic approximation of the transition state theory to calculate the frequency prefactor \( \tilde{\rho} \), and is given by\textsuperscript{22,23}

\[
\tilde{\rho} = \prod_{i} \frac{\nu_i^{\text{initial}}}{\nu_i^{\text{saddle}}},
\]

where \( \nu_i \) are the normal vibration modes (phonons). Note that the product in the denominator excludes the imaginary frequency corresponding to the unstable mode for the transition state. We determined the minimum-energy path for each migration and the corresponding migration energy barrier \( E_m \) using climbing-image nudged elastic band method.\textsuperscript{13,14} In NEB, a set of intermediate states (images) are distributed along the reaction path connecting known initial and final states. To ensure the continuity of the reaction path, the images are coupled with elastic forces and each intermediate state is fully relaxed in the hyperspace perpendicular to the reaction coordinate.

III. RESULTS AND DISCUSSION

To predict the effect of C concentration on Fe self-diffusivity for Fe-C alloys comprising a diverse PDC population, it is necessary to first predict the concentration of PDCs, and next determine the migration barriers for the statistically abundant PDCs. Here we detail the results of these calculations, which naturally require evaluation of multiple migration pathways.

A. Point-defect clusters: Formation and abundance

It has been established, both experimentally\textsuperscript{24} and theoretically\textsuperscript{6,11,17,25} that solute atoms of intermediate size (e.g., N, C, and O) in a bcc lattice are more stable at the octahedral sites. This introduces a tetragonal distortion to the neighboring Fe lattice.\textsuperscript{26} The relaxation of the nearest Fe atoms can be calculated as \( \Delta d_i \), where \( d_i \) is the \( i \)th nearest-
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TABLE I. Calculated binding energies ($E^b$) using DFT and MFS potential are in good agreement. The crystallographic directions (i,j,k) refer to the orientation of the Fe double vacancies. Binding energies for the clusters containing more than two carbon atoms per Fe vacancy are not shown here (see Ref. 6).

| Defect cluster | DFT  | MFS  |
|----------------|------|------|
| Va1C1          | 0.53 | 0.76 |
| Va1C2          | 1.46 | 1.73 |
| Va2(100)       | 0.18 | 0.21 |
| Va2C1(100)     | 1.05 | 1.05 |
| Va2C1(100)     | 1.84 | 1.65 |
| Va2C2(100)     | 2.31 | 2.25 |
| Va2C2(100)     | 3.20 | 3.13 |
| Va2(111)       | 0.12 | 0.14 |
| Va2C1(111)     | 0.80 | 0.93 |
| Va2C2(111)     | 1.93 | 1.61 |
| Va2C3(111)     | 2.57 | 2.39 |
| Va2C4(111)     | 3.62 | 3.09 |

neighbor (NN) solute-Fe distance before relaxation and $\Delta d_i = |d_{i1} - d_{i2}|$ is the change in distance due to relaxation. The calculated relaxations due to octahedral C in bcc Fe [$\Delta d_1 = 0.25\%$ (DFT) and 22% (MFS) and $\Delta d_2 = -1.5\%$ (DFT) and $-1.6\%$ (MFS)] are in good agreement with x-ray diffraction measurement$^{26}$ and previous theoretical calculation.$^{17}$ In contrast, while forming Va1C1 complex in the presence of Fe vacancy, the carbon atom relaxes from the octahedral position toward the vacancy and the calculated vacancy-carbon distances are 0.41$a_0$ (DFT) and 0.36$a_0$ (MFS), which are consistent with positron annihilation experiments.$^{27}$

Calculated binding energies $E^b$ for the PDCs are given in Table I. The results using our previously developed MFS potential$^{14}$ are in good agreement with DFT results, as expected from the DFT-based construction this MFS potential.$^{11}$ The binding energy for the Va1C1 complex, $E^b(Va_1C_1)$, is in good agreement with previous theoretical results.$^{17,28-30}$ In contrast, current experimental results range appreciably, 0.4 $\leq E^b(Va_1C_1) \leq 1.1$ eV, varying with experimental technique and sample preparation details.$^{31-34}$ The magnitude of $E^b(Va_1C_1)$ predicted from resistivity measurements on low-temperature irradiated pure and carbon-doped bcc Fe is comparatively high (1.1 eV).$^{34}$ Magnetic-after-effect measurement predicts a range of 0.1 $\leq E^b(Va_1C_1) \leq 0.65$ eV,$^{32}$ whereas positron annihilation data suggests an upper limit of 0.75 eV.$^{33}$ Calculated binding energy for the vacancy-carbon complex containing two carbon atoms, Va1C2, is in good agreement with previous DFT calculation.$^{17}$ In agreement with previous experimental findings,$^{34,35}$ it should be noted from Table I that it is energetically favorable to add a carbon atom to a pre-existing Va1C1 complex than to form a new Va1C1 complex.

It is observed experimentally that solute atoms in bcc Fe serve to increase the vacancy concentration well above the thermodynamically predicted levels in an otherwise perfect lattice.$^{5,7,8}$ Our previous DFT calculations showed that this is indeed the case for carbon-rich ferrite and found that Va1C2 complexes are the most abundant vacancy-containing defect cluster.$^6$ In general, we find that for any assumed vacancy concentration, the free monovacancy and divacancy concentrations decrease with increase in available carbon content in the solid solution (Fig. 1). Consequently, Va1C2 concentration increases due to comparatively high thermodynamic stability. The relative abundance of Va1C1, Va2C1, and Va2C2 clusters and all other larger vacancy-carbon PDCs containing more than two C atoms are found to be negligible.$^6$ Although Fig. 1 shows this trend for an assumed constant vacancy concentration of 10$^{-4}$ and at a specific temperature of 433 K (an annealing temperature for high carbon steel considered in our previous studies$^{6,11}$), this behavior is found to be generic for any vacancy concentration and temperatures below the Curie temperature at which bcc Fe transforms to fcc Fe.

B. Single-defect migration

Having determined the statistical abundance, we next calculate the migration barriers for PDCs corresponding to various possible migration pathways. Isolated point defects, Va1 and C1, are the smallest point defects, and thus the migration mechanisms are simpler compared to the large PDCs. Migration of a single vacancy in a bcc lattice has only two possible migration pathways: along the $\langle 100 \rangle$ and $\langle 111 \rangle$ (Fig. 2) directions. Migration along the $\langle 100 \rangle$ directions requires much higher energy of $\sim 3.5$ eV. In comparison, diffusion along the $\langle 111 \rangle$ direction requires minimum energy, and thus the minimum-energy pathway for vacancy diffusion. The calculated MFS-NEB migration barrier (0.84 eV) agrees well with the present first-principles DFT-NEB calculation (0.75 eV) and also with previous DFT calculations with different level of approximations (within $\sim 0.1$ eV).$^{36,37}$ However, the ex-
experimental (111)-migration barrier is widely scattered (0.55–1.28 eV) (Refs. 5, 38, and 39) and remains disputed despite numerous experimental attempts over decades.5,38–43 This wide experimental range is attributed to difficulties in preparing high purity bcc Fe, the high sensitivity of Fe vacancies to interstitial solutes, and also in the interpretation of experimental data.5,27,38–43

In contrast, the (octahedral) C solute atom has only a single expected migration pathway defined by the octahedral interstitial lattice. For such migration, it is intuitive to assume that the tetrahedral site connecting two neighboring octahedral sites would be the transition state (saddle point). The energy difference between the tetrahedral and octahedral state might be a good estimate of the activation energy for such migration.17,25 However, without this a priori assumption, we carried out NEB calculations for C migration in vacancy-free bcc Fe. We indeed find the tetrahedral carbon state as the saddle point. The calculated MFS migration energy is smaller (0.64 eV) than that predicted by DFT (0.86 eV), which is in good agreement with experimentally measured barriers ranging 0.81–0.88 eV.27,34,44,45

The solute (carbon) diffusion coefficient is a function of jump rate $\Gamma$, $D_c = \frac{1}{6} a_c^2 \Gamma$. Neglecting the formation entropy, $\Delta S_f$, the diffusion coefficient becomes, $D_c = D_0 \exp (-E_m/k_B T)$ with $D_0 = \frac{1}{6} a_c^2 \Gamma$. Here $a_c$ is the jump distance (between two nearest-neighbor octahedral sites in case of C diffusion) and $E_m$ is the corresponding C migration barrier. Within the Einstein approximation, we considered the vibrational modes of the atoms near the defect, which significantly reduced the size of the Hessian matrix to calculate phonon frequencies. The calculated prefactor $D_0(C) = 3.48 \times 10^{-7} m^2/s$ (DFT) and $4.72 \times 10^{-7} m^2/s$ (MFS) are in good agreement with previous theoretical predictions17,25 and the average experimental $D_o(C)$ below 350 K, 1.67 $\times 10^{-7} m^2/s$.46

C. Point-defect cluster migration

The number of possible migration mechanisms increases rapidly with defect cluster size and thus becomes increasingly expensive computationally for DFT-NEB approach. Instead we have used NEB-MFS to survey minimum-energy pathways for larger clusters. Although the Va$_2$C$_1$ cluster abundance in the bcc Fe-C alloy is negligible, we studied dissociative and associative migration pathways for this complex to compare with recent results.47 Schematic positions of vacancy and carbon for dissociative and full (associative) migrations for Va$_1$C$_1$ cluster.

- *TABLE II. Migration energy barrier for Va$_1$ diffusion in Va$_1$C$_1$ cluster. i→j (j→i) represents forward (backward) vacancy jump, as represented in Fig. 3(a). Present MFS-NEB results are compared with Ref. 47.*

| Vacancy jump | Present | Ref. 47 | Present | Ref. 47 |
|--------------|---------|---------|---------|---------|
| 1→2          | 0.59    | 0.49    | 0.18    | 0.28    |
| 1→9          | 1.84    | 1.60    | 0.87    | 0.64    |
| 2→3          | 0.98    | 0.79    | 0.83    | 0.61    |
| 2→4          | 0.81    | 0.63    | 0.60    | 0.47    |
| 2→5          | 1.30    | 0.95    | 0.96    | 0.73    |
| 2→6          | 0.81    | 0.63    | 0.60    | 0.47    |
| 2→7          | 0.86    | 0.63    | 0.67    | 0.50    |
| 2→8          | 0.86    | 0.68    | 0.65    | 0.48    |

- FIG. 2. (Color online) Monovacancy migration along (111) directions. The DFT and MFS results are in good agreement. Diffusion along (100) directions require much higher energy (∼3.5 eV) and is not shown. Filled square represents the Fe vacancy.

- FIG. 3. (Color online) Migration of the Va$_1$C$_1$ complex. Schematic representation of (a) dissociative and (b) associative migrations. Point defects are indicated by Arabic (Fe vacancy) and Roman (carbon) numerals. Position O(0,0,0) in (a) represents the origin. Similar conventions have been used throughout. (c) and (d) Migration pathways corresponding to the mechanisms described in (a) and (b), respectively.
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In contrast, migration of the $Va_2(111)$ cluster is more energetically expensive by $\sim 0.15$ eV. We have detailed the migration mechanisms and corresponding activation barriers of the $Va_2(111)$ in Appendix. Due to the lower concentration and higher barriers of $Va_2(111)$ migration, the concentration and migration barriers corresponding to $Va_2(100)$ were thus considered to represent the contribution of divacancy mobility in our subsequent calculation of Fe self-diffusivity as a function of C concentration.

The most prevalent Va-C complex at high carbon concentrations of interest is found to be the $Va_2C_2$ cluster (Fig. 1). Figures 5(a) and 5(b) summarizes different migration mechanisms and corresponding migration barriers of this complex. This PDC can migrate through dissociation into smaller PDCs ($Va_1C_2 \rightarrow Va + 2C$ indicating dissociation into three isolated point defects and $Va_1C_2 \rightarrow Va_1C_1 + C$ indicating dissociation into a smaller Va-C PDC and isolated carbon). These migrations are equally probable ($E_{\text{act}} = 2.28$ eV and 2.22 eV barriers, respectively). Further, these barriers are considerably higher than monovacancy and divacancy migrations, and thus these carbon-containing PDCs will not migrate and contribute appreciably to the self-diffusivity. Similar to the $Va_1C_1$ complex, the Va migration barriers toward

\begin{align*}
\Delta E_{\text{act}}^c &= 2.28 \text{ eV} \\
\Delta E_{\text{act}}^{c+i} &= 2.22 \text{ eV}
\end{align*}

In our subsequent calculation of Fe self-diffusivity, we consider the contribution of divacancy mobility in the $Va_2(100)$ and $Va_2(111)$ clusters.

FIG. 4. (Color online) Migration of the $Va_2(100)$ cluster. (a) Migration mechanisms via dissociative $Va_2(4NN)$ and associative $Va_2(111)$ intermediates. (b) Minimum-energy pathways for the mechanisms described in (a).
the carbon atoms for these dissociative $\text{Va}_1\text{C}_2$ migrations are much smaller (0.48 eV and 1.58 eV, respectively) than the corresponding $\text{Va}$ migration away from the carbon atoms [Fig. 5(b)], i.e., the reformation of the $\text{Va}_1\text{C}_2$ complex is more likely than the dissociation of this PDC. Although both the carbon and vacancy move together during the dissociative $\text{Va}_1\text{C}_2 \rightarrow \text{Va}_1\text{C}_1 + \text{C}$ migration, the analysis of the trajectories of both the $\text{Va}$ and $\text{C}$ [$\Delta$ in Fig. 5(b)] enables us to correlate the energetic peak (at reaction coordinate $\sim 0.15$) with dominant motion of the $\text{C}$, and the shoulder (at reaction coordinate $\sim 0.55$) with dominant motion of the $\text{Va}$. Compared to these dissociative migration mechanisms, migration of this PDC to maintain an intact $\text{Va}_1\text{C}_2$ cluster is less probable due to a much higher energy barrier [3.24 eV, Fig. 5(b)].

D. Self-diffusivity and carbon concentration

We found that monovacancies and divacancies are the only point-defect species with comparable migration energies. In contrast, the prevalent $\text{Va}_4\text{C}_2$ cluster is essentially immobile. Therefore, the diffusivity in ferritic or bcc Fe-C alloys is dominated by monovacancies and divacancies, and the Fe self-diffusivity can be written as a sum of these independent contributions,

\[
D^* = a_1^2 C_1 \nu e^{-(E_m/k_B T) } f_1 + 2 a_2^2 C_2(100) \nu e^{-(E_m/k_B T) } f_2,
\]

(7)

Here $a$ is the jump distance calculated from the atomic configurations, $C$ are the PDC species concentrations, $\nu$ is the jump frequency [calculated from Eq. (6)], $f$ is the jump correlation factor, and $E_m$ are the migration barriers. The frequencies and energy barriers for $\text{Va}_1(100) \rightarrow \text{Va}_1(4\text{NN}) \rightarrow \text{Va}_2(100)$ and $\text{Va}_2(100) \rightarrow \text{Va}_2(111) \rightarrow \text{Va}_3(100)$ migrations [Fig. 4] are labeled with I and II, respectively. The correlation factor for divacancies, $f_2$, depends on the ratio of the jump frequencies, $P_2/P_1$, corresponding to these mechanisms, for which we used the functional form introduced by Belova et al. through Monte Carlo simulations. We find that self-diffusivity decreases rapidly with increasing carbon content in the matrix (Fig. 6) and further that this general conclusion is independent of temperature (below 1180 K at which bcc Fe is the stable crystal structure). The reason for such decrease in $D^*$ with increasing carbon concentration is straightforward in terms of relative mobility of the PDC spectrum. First, the concentrations of mobile $\text{Va}_1$ and $\text{Va}_2$ decrease with increasing carbon content. Second, with increasing carbon content, the concentration of $\text{Va}_4\text{C}_2$ PDCs also increases. These PDCs are effectively immobile due to their high migration barrier determined above. Given our finding that diffusivity is dominated by monovacancies and divacancies, we proceed with the hypothesis that diffusivity vanishes with decreasing vacancy concentration [see Eq. (7)]. Thus, as Fig. 6 shows, $D^*$ vanishes when there are no longer free (monovacancies or divacancies) vacancies present. This vanishing free vacancy population and diffusivity naturally occurs when the carbon concentration is twice the vacancy concentration, at which point all free vacancies are subsumed into $\text{Va}_4\text{C}_2$ clusters.

Note that the present calculations consider the concentration of carbon in the bcc Fe matrix or lattice, and thus lattice diffusion. Clearly, the total carbon concentration in such alloys also includes finite carbon concentration at dislocations, grain boundaries, and other phases. However, the overall trend of $D^*$ with carbon concentration in the lattice is preserved in such cases of additional complexities. The present result for bcc Fe, wherein self-diffusivity decreases with increasing carbon content (Fig. 6) contrasts with the experimentally observed carbon dependence in the other high-temperature polymorph, (austenitic) fcc Fe-C. Carbon enhances self-diffusivity by nearly one order of magnitude upon addition of 1 wt % C in the fcc Fe matrix. This enhancement is due to the fact that in fcc Fe, the smallest vacancy-carbon complex, namely, the $\text{Va}_4\text{C}_1$ cluster, contributes considerably to self-diffusion. In contrast, for bcc Fe we find this complex is not statistically abundant in the PDC spectrum, and thus does not contribute to self-diffusion. Similar carbon concentration dependent decreases in self-diffusion for bcc Fe have been reported by Fu et al. However, that previous theoretical approach relied on an important a priori assumption that all PDCs containing C are immobile, assumed the atomic jump frequency, and neglected the contribution of divacancy clusters to self-diffusion. There is no explicit experimental measurement of Fe self-diffusivity as a function of carbon content in bcc Fe, our results on dislocation-free single-crystal bcc Fe are consistent with the experimental observations of macroscopic creep strain rates (at a given applied stress and temperature in the power-law creep regime) decreasing with increasing total carbon content in polycrystalline Fe-C alloys. This observed creep operates by a dislocation climb mechanism (in accordance with its power-law stress dependence) as well as the result, is related to diffusivity of lattice vacancies. Although the quantitative measurement of the differential distribution of carbon in different regions of single-crystalline and polycrystalline alloys is a current experimental challenge, it is quite plausible that the carbon concentration...
within the bcc Fe matrix increases with increasing total carbon content in bcc Fe-C alloy.) The present methodology and results thus present a quantitative dependence of self-diffusion on interstitial solute in (single-crystalline) structural alloys, and is also consistent with the qualitative solute dependence of steady-state creep in polycrystalline alloys.\footnote{15}

\section*{IV. SUMMARY}

Here, we employed atomistic simulations combined with statistical theory to study the abundance and microscopic migration mechanism of various point-defect clusters in ferritic (bcc) Fe-C alloys. For this system of interest, we find that self-diffusivity within dislocation-free regions of the matrix decreases with increasing carbon content and becomes negligible as carbon concentration in the lattice approaches zero. Indeed, the activation barrier for the most probable migration of divacancy in this configuration is \(\Delta E = 0.15\) eV higher than in the \(\langle 100\rangle\) counterpart.\footnote{15}\footnote{15} We gratefully acknowledge initial financial support from SKF Global, Inc., helpful discussions with J. Slycke and B. Hosseinkhani, U.S. National Defense Science and Engineering Graduate Program (T.T.L.), and the U.S. AFOSR PECASE (K.J.V.V.).

\section*{ACKNOWLEDGMENTS}

Although the divacancy cluster in the \(\langle 111\rangle\) configuration is not statistically abundant in the PDC spectrum, we calculated the corresponding migration barriers to compare with the \(\langle 100\rangle\) counterpart. Indeed, the activation barrier for the most probable migration of divacancy in this configuration is \(\Delta E = 0.15\) eV higher than in the \(\langle 100\rangle\) configuration. Different possible migration paths for \(\text{Va}_2\langle 111\rangle\) are shown in Figs. 7(a)–7(h). A \(\text{Va}_2\langle 111\rangle\) cluster [Fig. 7(e)] can migrate to a \(70.5^\circ\)-rotated \(\text{Va}_2\langle 111\rangle\) configuration [Figs. 7(c) and 7(g)] in two possible paths via different intermediate states of varying NN distances. The energy cost for the migration through a \(\text{Va}_2\langle 3\text{NN}\rangle\) intermediate [Fig. 7(b)] is \(\Delta E = 0.20\) eV higher compared to the migration via \(\text{Va}_2\langle 100\rangle\) configuration [Fig. 7(f)]. Corresponding migration barriers are shown in Fig. 7(i). A \(\text{Va}_2\langle 111\rangle\) cluster [Fig. 7(e)] can also completely

\begin{figure}[h]
\centering
\includegraphics[width=\columnwidth]{figure7}
\caption{(Color online) Migration of the \(\text{Va}_2\langle 111\rangle\) cluster. (a)–(h) Different migration mechanisms through various intermediates. (i) Minimum-energy pathways of \(\text{Va}_2\langle 111\rangle\) migration to a final \(70.5^\circ\)-rotated \(\text{Va}_2\langle 111\rangle\) cluster through the dissociative \(\text{Va}_2\langle 3\text{NN}\rangle\) and associative \(\text{Va}_2\langle 100\rangle\) intermediates.}
\end{figure}
diffuse to (111) direction [Figs. 7(a) and 7(b)] through one of these \((\text{V}_\text{A}_2, 3\text{NN})\) or \((\text{V}_\text{A}_2, 100))\) intermediates. We see the migration barriers depend only on the intermediates (dissociated (3NN) or bound (100) cluster) and are independent of the final divacancy configurations, i.e., whether it is rotated by \(70.5^\circ\) or diffuses only along (111) direction.

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