Consequences of Styrene monomer leakage into the atmosphere- a damage to the human beings

Abstract

India has been facing many problems in recent times due to pandemic Covid-19, pollution, population and also many other natural disasters. In addition to these problems, the country is also facing some tragedies such as leakage of gases from industries and blasts. Many gas leakages and blasts were seen in India from the last few decades. Recently, the Visakhapatnam in Andhra Pradesh, India also experienced a gas leakage into atmosphere, which is a dangerous chemical gas i.e styrene from Polymer Company named LG polymer India Company limited on May 7th 2020. LG polymers manufacture plastics and expandable plastics using styrene monomer as raw material. Many people were affected and died due to this leakage of dangerous chemical gas which was released from the polymer industry at early hours of morning. The gas leakage has spread to many nearby villages and has spread to many kilometres. Mainly respiratory problems, eye irritation and skin irritation are the symptoms of the exposure to styrene. Long exposure to this chemical gas may cause the respiratory diseases that effect during lifetime. There should be some environmental measurements and remedies should be taken by the government and also by the people in order to overcome these types of situations.

Keywords: consequences, leakage, polystyrene, styrene monomer

Introduction

Visakhapatnam is a fast-growing metropolitan city that has more attracted by industries due to facilities to establish their values. Industrial growth in Visakhapatnam city is noticeable to urban agglomeration in population and the whole city is at a 10km distance from the Bay of Bengal shore. Due to various industries like petroleum refineries, steel, zinc, and fertilizer plants, polymer, cement factories, dairies there was a drastic deterioration of air, water quality. The industrial emissions have a combined deteriorating effect on the human health and well-being of the local communities, and mainly weaker sections are more disposed to these adverse impacts.

Recently the poisonous gas namely styrene was leaked on 07th May 2020 in Visakhapatnam, Andhra Pradesh (Figure 1(a)). So far, 12 individuals have died and people in hundreds number were infected. The tragedy in Vishakhapatnam shows its impact on five kilometer surroundings and affecting people in thousands. Previously, in India, massive chemical gas (Methyl-Isocyanate, MIC) tragedy happened at Bhopal in 1984 and 8000 people died in the first week and 5lakhs of the population were affected by the MIC gas. And another gas leak was on 2014, June 27th was the explosion of GAIL pipeline, an enormous fire broke out succeeding a heavy blast in Gas Authority of India Limited (GAIL) underground gas Pipeline i.e about 18” size at Nagaram, East Godavari district which is in Andhra Pradesh, India in which about 15 people have died and many got injured. There are several other gas leaks in which many people were injured and many people died. Some of the leaks are Bhilai steel plant gas outflow in the year 2014, Delhi gas leakage in the year 2017.

Similarly, the styrene gas tragedy in Visakhapatnam shows the leakage of styrene gas from LG Polymer India Private Limited (LGPI) as shown in Figure 1(b). LGPI is located in the RRV Puram which is part of Visakhapatnam and leads in the development, manufacture and services in the field of polymers. Presently LGPI is one of the prominent and leading manufacturers of Polystyrene as well as expandable Polystyrene in India.

Figure 1 The styrene gas tragedy in Visakhapatnam shows the leakage of styrene gas from LG Polymer India Private Limited (LGPI).
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Preparation of styrene

For the complete production of Styrene involves three steps i.e.

a) Synthesis of ethyl benzene: Ethyl benzene is prepared by the reaction of ethylene and benzene in the presence of a Friedel–crafts alkylation catalyst such as Aluminium Chloride at about 95°C (7)

\[
\text{Benzene} + \text{CH}_2=\text{CH}_2 \rightarrow \text{Ethyl benzene}
\]

b) Ethyl benzene dehydrogenation: By the process of dehydrogenation of the ethyl benzene crude Styrene is formed (7)

\[
\text{Ethylbenzene} \rightarrow \text{Styrene} + \text{H}_2
\]

c) Purification of styrene: By the process of dehydrogenation, crude styrene is formed. The obtained product consists of nearly 61% ethyl benzene and 37% styrene & 2% of aromatic hydrocarbons such as benzene. As the boiling point of the styrene is very high (145.2°C), the purification is very difficult.8

What happens in atmosphere if Styrene leak?

Styrene is a volatile organic compound and it is a liquid with a high boiling point. Air pollution modeller Awkash Kumar said that “Styrene in the air on 7 may 2020 when the leak was occurring could have been over 20 ppm up to 2km of the LGPI, assuming an hour’s leak and affected about 5 villages (R.R.Venkatapuram, Padmapuram, BC Colony, Gopalapatnam and kampara palem) around. Significantly high and more levels of pollutant dispersal could have led people to fall unconscious,”.

Besides, the real-time ambient air assessment and monitoring stations network of the Central Pollution Control Board (CPCB) monitors three VOCs such as Toluene (C7H8), Benzene (C6H6), Xylene (C8H10) in Visakhapatnam, 14km downstream of the spot where the leakage of gas incident happened and the results shown xylene levels up to 18ppb, toluene levels up to 35ppb and benzene levels up to 12ppb (Figure 2). After gas leakage it may spread to 5km surroundings places which have a dense population as seen in Figure 2. Once vaporises the styrene in atmosphere, it may combine or react with other VOCs and travel long range. These vapours may fall on the water bodies of rivers or lakes which are the great sink for drinking water, more chances to pollute.

Health effects of Styrene on human beings

According to the United States Environmental Protection Agency (USEPA), short-term contact with low levels of styrene in humans will irritate to the mucous membrane and eyes, and also gastrointestinal effects. On the other side, long-term exposure and contact with styrene results in severe effects on the Central Nervous System (CNS), leading to fatigue, headaches, weakness, loss of hearing, nerve damage and also depression.9 According to the Health and Safety Executive, a UK government agency is responsible for regulation and administration of work-place health, safety, and welfare, this chemical can cause loss of consciousness and even death at high exposure levels.

Figure 2 The benzene levels in ppb shown in Vishakhapatnam and Amaravati.

Styrene is toxic but it gives a severe effect on humans by ingestion or inhalation. Acute (short-term) exposure or contact to the styrene by humans may cause respiratory problems, eye irritation and also gastrointestinal effects.4 Styrene gas effects mainly on the brain cause the feeling of drowsiness and drunkenness, color vision defects, tiredness, and confusion. The cause of death in animals and humans is possibly asphyxia–oxygen deprivation. Aside from its effects on humans, the styrene gas also affects the life of birds, cattle, rats and dogs.5 Dermal contact may cause irritation, itching and erythematous popular dermatitis. As styrene is absorbed through the skin, systemic
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The limited evidence for cancer from styrene in humans is from occupational studies showing increased risks for lymphohematopoietic cancers, such as leukemia and lymphoma, and genetic damage in the white blood cells, or lymphocytes, of workers exposed to styrene. There is also some evidence for increased risk of cancer in the pancreas or esophagus among some styrene workers, but the evidence is weaker than that for lymphohematopoietic cancers.

Exactly how styrene causes cancer is not fully understood, but styrene is converted, in laboratory animals and humans, to styrene-7,8-oxide, which is listed in the Report on Carcinogens as reasonably anticipated to be a human carcinogen. Styrene-7,8-oxide causes genetic damage and has been found in the blood of workers exposed to styrene.

Suggestions or policies implementation

It is very highly reactive and it will polymerize rapidly with a violent explosive reaction. This demands transportation, proper handling and storage by adding polymerization inhibitors in sufficient quantities throughout these operations. Besides, an environmental information system needs to be adopted and regulate the protection rules of environment. Policies need to implement to minimize the risk factor.10,11

Conclusion

Chemical gas tragedy happened in Visakhapatnam and people in thousands were affected and 12 were died due to leakage of styrene gas into atmosphere. In general, styrene is a toxic gas as it is inhaled by humans and causes respiratory issues. Also, it may spread to nearby locations and pollute the environment predominantly air and water. Hence, the proper care should be taken to decrease or reduce the risk factor. The government should take precautionary measures to decrease the risk of these types of leakages and also to decrease the risk factor to humans.

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Conflicts of interest

The authors declare there are no conflicts of interest.

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