Fish canning industry wastewater treatment for water reuse – a case study

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A B S T R A C T

The valorization of wastewaters from the fish canning industry is of great concern, not only because of the high quantities generated, but also economic and environmental benefits may result from a proper treatment approach of the waste generated while reducing costs related to wastewater discharge.

A limiting factor for reuse and recycling treated fish canning wastewater into an industrial plant and also for other uses is the high salt content, which persists even after conventional treatment. So, the reuse of fish canning industrial wastewater was assessed by combining conventional treatments, such as sedimentation, chemical coagulation-flocculation and aerobic biological degradation (activated sludge process) followed by a polishing step by reverse osmosis (RO) and ultraviolet (UV) disinfection.

In this investigation all these processes were optimized in order to remove essentially the effluent suspended particles (primary treatment), the organic matter content in the biological aerated reactor (secondary treatment) and, finally, the remaining salts and microorganisms (tertiary treatment).

The overall removal efficiencies obtained were: 99.9% for dissolved organic carbon (DOC), 99.8% for oil and grease (O&G), 98.4% for total suspended solids (TSS), above 96% for anions and cations and 100% for heterotrophic bacteria expressed as colony-forming units (CFU). The final clarified effluent was found to have the quality requirements to be recycled or reused in the industrial plant, allowing the reduction of the effluent to be discharged, the water use and the costs of tap water for industrial use.

As regards the energy and chemicals costs, to obtain a treated effluent to be reused in the process costs 0.85 €/m³. This value can be reduced by about 60% if the goal is only to meet the legislated standards for the effluent discharge into water bodies. Tap water for the industrial plant costs about 2.1 €/m³.

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1. Introduction

Water consumption in fish processing industry and the production of high strength wastewaters are of great concern worldwide (Chowdhury et al., 2010). In fact, in this type of industries a huge amount of water is used throughout all steps, including cleaning, cooking, cooling, sanitization and floor washing. Fish canning wastewaters are known to contain organic contaminants in soluble, colloidal and particulate form (Chowdhury et al., 2010). The biodegradable organic matter is mainly in the form of proteins and lipids. The concentration and volume of wastewater from fish processing varies widely, depending on the fish to be processed, the additives used (e.g. brine, oil, tomato sauce), the unit processes involved and the source of the water (Palenzuela, 1999), among other factors. There is a need, for both economic and environmental sustainability reasons, to consider the wastewater treatment in order to obtain water with the quality requirements established by the European Directive 98/83/EC (Table 1), which allows its reuse into the industrial process.

Fish processing wastewaters are generally treated using physical-chemical methods (Fahim et al., 2001), biological methods (Palenzuela, 1999), or a combination of both. In integrated treatment systems, sedimentation, dissolved air flotation and pH adjustment are usually employed as primary treatment steps (Zufía and Aurrekoetxea, 2002) and the biological processes (aerobic and anaerobic) are known to be the more appropriate for organics
removal (Alexandre et al., 2011; Chowdhury et al., 2010). However, fish canning wastewaters also have high salinity. The elimination of salts is usually expensive and, on the other hand, the high salinity and the seasonal variation of the effluent characteristics make difficult to remove the organic matter by a biological process (Zufa and Aurrekoetxea, 2002). Nevertheless, for water reuse, a polishing step to remove the effluent salinity is necessary. Recent studies revealed that membrane separations may help in solving the problem of attaining a quality of water that allows being recycled back to the process (Mavrov and Belieres, 2000). Membrane separation allows the removal of contaminants, the recovery of dissolved or suspended high added value constituents (proteins, pigments, enzymes, etc.) (Martin et al., 1995) and even the water reuse for certain applications (Rautenbach and Linn, 1996). Kuca and Szaniawska (2009) studied the application of microfiltration using ceramic membranes for treatment of salted aequous effluents from fish processing and Pérez-Gámez et al. (2011) investigated the performance of three ceramic membranes in the treatment of the press liquor resulting from the compaction of sardine by-products. However, reverse osmosis is based on a lower pore size (around 0.001 micron), which allows the retention of smaller particles, including most ions present in the water. Reverse osmosis has been shown to be able to separate the salt content out of wastewater, retaining small ions such as Na⁺ or Cl⁻ (Rautenbach and Mellis, 1995).

Although conventional treatment processes remove up to 95–99% of some microorganisms, the presence of pathogens render the water inappropriate for direct reuse. The safe reuse of water therefore depends on wastewater disinfection (Gómez et al., 2006). Requirements for wastewater reuse are based principally on biological quality considerations. However, due to wide fluctuations in industrial effluent quality, this becomes more challenging. Chlorination is the most widely used method to inactivate pathogenic microorganisms in water and wastewater. However, questions have been raised regarding the formation of disinfection by-products (DBPs), known as carcinogens (Liberti et al., 2003), the erosion of water pipes and unsafe operation (Guo et al., 2011). In recent years, UV light has been increasingly used as a disinfectant, to inactivate bacteria without the formation of toxic by-products (Hijnen et al., 2006). In addition, the photoreactor design and lamp technology have improved considerably (Cassano et al., 1995). These facts led to the recognition by US Environmental Protection Agency (USEPA) that UV disinfection is the best current disinfection technology (Hijnen et al., 2006). UV light, emitted by mercury arc-lamps (low or medium pressure), is effective against a variety of pathogens including viruses, bacteria and protozoan cysts (Hallmich and Gehr, 2010). Essentially, the inactivating effects of UV rays are due to DNA changes (e.g., strand breaks, formation of thymine dimers) that result in inhibition of replication and, in case of lethal doses, in a loss of reproducibility. A disadvantage of this method is the possible reactivation of UV damaged microorganisms. However, reactivation decreases significantly when the water is exposed to visible light, simultaneously or prior to UV irradiation (Guo et al., 2009; Hallmich and Gehr, 2010).

This work intends to develop a real, effective and workable solution that avoids and/or reduces the rejection of wastewaters from the fish canning industrializing stream in the most appropriate way to obtain a product with higher added value (water) which can be reused. This objective will be achieved through the optimization of a series of various treatment steps envisaging the wastewater reuse in the industrial process, while reducing the discharge of effluent, limiting the water use and saving money. To our knowledge this is the first investigation of a combined treatment of fish canning industrial wastewaters for water reuse.

2. Fish canning process

The fish canning production process begins with the reception of raw materials (fish, sauces and packaging materials). In the fishing boats, the fish is kept in water tanks. At the factory, the water drains out and the fish is weighed and stored again in salted water. The main waste stream resulting from this operation is the water used in the fish transport, along with water used for preservation in the boats. The wastewater from this stage contains blood, fish, rocks and sand from the fishing boats tanks.

Then the fish is placed in brine, a process that requires water and produces an effluent especially rich in salt, blood and scales. Then unwanted fish parts are removed, particularly head and viscera, to which is also necessary the use of water. From this process results an effluent mainly contaminated with salt, blood and fish waste that cannot be used as food. In the fish washing process water is used in abundance. From this operation results a wastewater containing mainly blood, oil, flakes, salt and fish tissues.

The following process corresponds to the cooking and subsequent cooling of the fish. The fish cooking is done through direct contact with steam. After cooking, the water produced is drained and the fish is cooled by water spraying. There are two sources for the wastewater generated in this process: one originated from the cooking, which has a very high concentration of organic material and fats and other from washing and cooling (usually seawater) that has low organic load, but high salinity. Hence, the fish goes to canning, sauces addition, spiking and cans washing processes. This final wash leads also to an effluent that contains, essentially, fish fat. Finally, the cans are sterilized with steam and cooled with water, which also results in a wastewater. In addition to all these wastewaters produced, liquid spills occur due to shipping, handling, canning and cleaning of equipment, involving waste of raw materials.

| Parameter                | Unit | Parametric value |
|-------------------------|------|------------------|
| Aluminum                | mg/Al/L | 200               |
| Ammonium                | mg NH₄/L | 0.50             |
| Calcium                 | mgCa/L | –                |
| Chlorides               | mg Cl/L | 250              |
| Clostridium perfringens | N/100 mL | 0                |
| Color                   | mg P°C/L | 20               |
| Conductivity            | µS/cm, 20 °C | 2500       |
| Total hardness          | mg CaCO₃/L | –            |
| pH                      | pH units | 6.5 ≤ 9          |
| Iron                    | mg Fe/L | 200              |
| Magnesium               | mg Mg/L | –                |
| Manganese               | mg Mn/L | 50               |
| Microcystin — full LR  | µg/L | 1                |
| Smell, 25 °C            | Dilution factor | 3         |
| Oxidability             | mg O₂/L | 5                |
| Sulphates               | mg SO₄/L | 250              |
| Sodium                  | mg Na/L | 200              |
| Flavor, 25 °C           | Dilution factor | 3         |
| Colony-forming units    | N/mL, 22 °C | Without abnormal change |
| Coliform bacteria       | N/100 mL | 0                |
| Total organic carbon (TOC) | mg/L | Without abnormal change |
| Turbidity               | UNT | 4                |
| α-total                 | Bq/L | 0.5              |
| β-total                 | Bq/L | 1                |
| Tritium                 | Bq/L | 100              |
| Total indicative dose   | mSv/year | 0.10            |
| Residual disinfectant   | mg/L | –                |
The water used in the various steps of the process can be replaced by reused water from the treatment of the wastewaters generated.

3. Materials and methods

3.1. Wastewater characterization

A sampling program was carried out in a selected fish canning company with a production capacity of 100,000 cans/day, located at Póvoa de Varzim, in northern Portugal. Several samples of 100 L of wastewater were collected weekly during 3 months and characterized after a preliminary screening stage. Standard Methods for the Examination of Water and Wastewater (APHA, 2005) were adopted for the measurement of total suspended solids (TSS), volatile suspended solids (VSS), dissolved organic carbon (DOC), chemical oxygen demand (COD), biological oxygen demand (BOD5), oil and grease (O&G), total phosphorus (Ptotal), total soluble nitrogen (Ntotal soluble), ammoniacal nitrogen (Nammoniacal) and several anions and cations. A Shimadzu 5000A Total Organic Carbon analyzer was employed for DOC measurements. The reported values represent the average of at least two measurements; in most cases each sample was injected three times, validation being performed by the apparatus only if the coefficient of variation (CV) was smaller than 2%.

The pH was measured using a selective electrode (Hanna Instruments HI 1230) and a pH meter (Hanna instruments HI 8424) and the conductivity at 20 °C was determined using a conductivity probe (WTW TetraCon 325) and a conductivity meter (WTW LF538).

Anions were measured by ion chromatography (Dionex ICS-2100) using a Dionex Ionpac (column AS 11-HC 4 × 250 mm; suppressor ASRS 300 4 mm). Cations were analyzed also by ion chromatography (Dionex DX-120), using a Dionex Ionpac (column CS12A 4 × 250 mm; suppressor CSRS 300 4 mm). Isocratic elution was achieved with NaOH 30 mM/methanesulfonic acid 20 mM at a flow rate of 1.5/1.0 mL/min for anions/cations analysis, respectively.

3.2. Wastewater treatment sequence

A sequential treatment for fish canning wastewaters was investigated for water reuse (Fig. 1). This sequence consists of sedimentation/flotation (gravity separation) for settleable solids, floatable oils and grease removal, coagulation-flocculation using FeCl3 as coagulant for additional removal of oil and grease and non-settleable solids, aerobic biological treatment (conventional activated sludge process) for soluble organic matter degradation, reverse osmosis (low pressure composite membrane) for salts removal and, finally, UV disinfection (low pressure mercury lamp) for microorganisms' inactivation.

3.2.1. Sedimentation/flotation

The first stage of the wastewater treatment process was sedimentation/flotation. The removal of suspended particles by sedimentation or flotation depends upon the size and specific gravity of the particles. Suspended solids may remain in suspension if their specific gravity is similar to water while very dense particles settle down. So, the effluent was left in graduated cylinders (42 cm height and 5.8 cm internal diameter) for 1.5 h. The optimum sedimentation time was determined in previous sedimentation tests. The graduated cylinders are provided with a sampling port, 5 cm above the bottom, which allows taking samples directly from the middle layer (Fig. 2(a)). These samples were then analyzed for TSS, DOC and O&G.

3.2.2. Coagulation/flocculation

The coagulation/flocculation process is used to remove colloidal particles and, whenever the natural settling of suspended material is not possible or is too slow, to provide effective clarification. The addition of coagulants causes the destabilization of the colloids,
followed by particles collision and flocs formation. Flocs are separated from the water by sedimentation or flotation. The performance of a particular coagulant depends upon the quality of the wastewater. Different types of coagulants were selected and their performance was assessed by varying the dosages employed and the pH of the wastewater.

So, the clarified wastewater, from the sedimentation/flotation stage, was submitted to chemical coagulation/flocculation, using a standard jar test apparatus (Jar tester JLT6, VELP Scientifica) (Fig. 2(b)). The adequate coagulant dosage and pH were optimized by performing several jar-tests, using 7 coagulants (2 organic compounds (RIPOL 070 and RIPOL 1815) and 5 inorganic salts (aluminum sulfate ($\text{Al}_2\text{(SO}_4\text{)}_3\cdot\text{16H}_2\text{O}$), ferric sulfate ($\text{Fe}_2\text{(SO}_4\text{)}_3$), ferric chloride ($\text{FeCl}_3$), calcium chloride ($\text{CaCl}_2$) and polyaluminium chloride (PAX-18)), at 3 out of 5 different coagulant dosages (from 20 to 400 mg/L). The pH was varied between 5 and 9 for each optimum coagulant dosage achieved. Each jar was filled with 500 mL of sample and the coagulant dose and pH value were adjusted to the intended values. The experimental procedure consisted of rapid mixing at 150 rpm for 3 min and, after that, the wastewater was moderately stirred at 20 rpm for 15 min to promote flocculation. Finally, a 1 h sedimentation stage allowed the flocs formed to settle. The supernatants obtained were then characterized in terms of TSS and O&G.

3.2.3. Biological treatment

The biological treatment was applied to the fish canning wastewater after sedimentation/flotation and coagulation/flocculation steps in order to evaluate the organic matter removal efficiency by activated sludge. A sample of suspended biomass from the aeration tank of a Municipal Wastewater Treatment Plant (Porto, Portugal) was used as inoculum. Aeration of the wastewater provides oxygen to aerobic microorganisms’ metabolic process, causing the flocs formation – activated sludge – and allowing the reduction of the organic content of the wastewater. The flocs are separated from the treated effluent by sedimentation.

The experiments for this study were performed in a biological system that consists of a 110 L feed tank containing the wastewater to be treated, an aeration tank (internal diameter ($\text{ID}$) = 19 cm, height ($H$) = 33 cm, working volume ($V$) = 6 L) equipped with air diffusers at the bottom to ensure the oxygen supply and the mixing of the whole reactor content (the air flow rate was about 6 L min$^{-1}$), a secondary sedimentation tank ($\text{ID}$ = 19 cm, $H_{\text{cylinder}}$ = 31 cm, $H_{\text{conic}}$ = 15 cm, $V$ = 6 L) equipped with a sludge recirculation system.
and a storage tank for the final effluent (treated wastewater). There are also two peristaltic pumps operating at adjustable flow rate: one for reactor feeding and another for sludge recirculation (Fig. 3(a)).

Before conducting the biological treatment studies themselves, the acclimatization of the inoculum was carried out in batch mode during 20 days in order to obtain a biomass concentration in the reactor greater than 1500 mg SSV/L. Once reached this value, the operation was shifted to continuous mode. Temperature (T), pH and dissolved oxygen (DO) were approximately $T = 22 \pm 2 \, ^\circ\text{C}$, $\text{pH} = 7.5 \pm 0.5$, $\text{DO} = 2.0 \pm 0.5$ mg/L, respectively, throughout the experiments.

The reactor was operated at different hydraulic retention times (HRT): 5.0, 6.0 and 8.0 h, and the DOC of the feeding stream was between 600 and 700 mg/L, approximately. Influent and effluent samples were taken to measure DOC values.

3.2.4. Filtration

The sand filter allows the particles retention from the upstream treatments, as well as of the biomass from the secondary clarifier. The wastewater passes downstream through several sand layers of decreasing particle size, which allows for the retention of the impurities.

Paterniani and Conceição (2001) claim that sand filtration process has some advantages over other technologies, highlighting the non-chemicals usage, not requiring process control sophisticated equipment, besides being of simple construction. These advantages, together with the use of alternative materials and simple hand work, can reduce the initial costs of construction, operation and maintenance.

So, in order to remove residual suspended solids found in the secondary effluent that may interfere with the subsequent RO process, simultaneously reducing the concentration of organic matter and turbidity (Hamoda et al., 2004), the biologically treated effluent was passed through a rapid sand filter, before reverse osmosis and UV disinfection processes.

The filter used in the filtration stage was a gravity sand filter (50 cm height and 11.3 internal diameter) composed of a 40 cm layer of sand (density 2.6 g/cm$^3$) with diameter between 0.5 and 1 mm. Two 5 cm gravel layers were placed on the top and on the bottom of the filter column to better distribute the water over the sand and to act as support medium, respectively (Fig. 3(b)). The filter was designed and operated to provide an average filtration rate of 2.4 m$^3$ m$^{-2}$ d$^{-1}$. The efficiency of the sand filter was assessed in terms of TSS reduction.

3.2.5. Reverse osmosis

Reverse osmosis is a membrane separation process, under pressure, in order to produce very high quality water, by removing some suspended contaminants, and separating low molecular weight dissolved substances (ions and soluble organic and inorganic compounds), and microorganisms.

The effluent from the sand filter was stored in a 600 L-capacity feed tank and then pumped to the reverse osmosis system by a MEW pump QB-70, at a maximum flow rate of 7.7 L/min.

The reverse osmosis pilot consisted of an RO 250 Model from AQUAQUIMICA, Lda. (Fig. 4), allowing a maximum permeate flow rate of 250 L/h and a pressure between 7 and 15 bar. This system consists of one Thin Layer Composite (TLC) membrane (5 µm, 40 × 40 inch), membrane housing of plastic reinforced with fiberglass (PRFG) and works at operating temperatures between 13 and 30 °C. The installed pump is Efaflu/Lowara with a 2.2 kW/400 V motor, minimum pressure inlet alarm and pump protection.

The reverse osmosis (RO) tests were carried out at permeate flow rates between 192 and 216 L/h and concentrate flow rates between 60 and 270 L/h. The pressure was varied between 8.5 and 10.0 bar. Feed flow rate and operational pressure were adjusted by controlling valves of recirculation and concentrated streams and visualized in pressure gages and flow meters. The RO plant was operated at a recovery rate between 50 and 75% by adjusting the concentrate drain valve according to the permeate flow achieved.

The permeate is sent to the UV disinfection unit and the concentrate is discharged into the sewerage system. Before sending the permeate to the last treatment step some samples were collected from time to time to be analyzed for DOC, N$_{\text{total}}$ soluble conductivity, anions and cations.

The performance of RO membranes tends to decrease during operation as a biomass layer builds up on the membrane surface, which reduces the flux and permeate quality with a concurrent increase in differential pressure. Blockage of the membranes may lead to precipitation of salts on the membrane surface and in the

Fig. 4. Pilot plant for reverse osmosis studies.
feed inlets. A specific cleaning regime was developed to prevent the formation of the biomass layer, and consequently formation and fouling, by passing chlorine solution (1 mg/L) throughout the system.

3.2.6. UV disinfection

The UV disinfection system uses UV light at sufficiently short wavelength to destroy the organism’s genetic material (DNA and RNA), leaving them unable to perform vital cellular functions, as reproduction.

The UV disinfection equipment is composed of one 45 W low-pressure mercury discharge lamp, and 5 µm filters, an AISI 304 stainless steel chamber, an hour meter and electronic chokes. All components of the system are inside a steel AISI 316 L cylindrical reactor (Fig. 5). The equipment shall provide a minimum UV radiation dose of 25 mW s⁻¹ cm⁻² at the end of the lamp effective life.

The clarified effluent from the reverse osmosis system is fed to the UV disinfection system at a flow rate of 250 L/h. Then samples are collected after the disinfection and, finally, this effluent is evaluated as regards its suitability for water reuse. The samples before and after being subjected to UV disinfection were analyzed for the presence of microorganisms, by performing heterotrophic bacteria counts by scattering method with AGAR medium plates. The plates were incubated at 22 and 37 °C, during 24 and 48 h. Several dilutions were made for each sample.

4. Results and discussion

4.1. Analysis of wastewaters from fish canning industries

The quality of fish canning wastewaters varies according to the overall production of the fish canning industry. In order to obtain a representative set of data on effluent properties, several samples were collected at different times and analyzed. Their characteristics are presented in Table 2, where maximum and minimum values obtained for several samples are reported. The high BOD₅ and COD values indicate a heavy contamination by organic matter. High concentrations and conductivity (corresponding to salinity between 2.5 and 15 ppt).

4.2. Primary treatment by sedimentation/flocculation and coagulation/flocculation

The sedimentation/flocculation and the coagulation/flocculation processes are the most important physico-chemical treatment steps in the primary treatment of some industrial wastewaters to reduce the suspended and colloidal particles responsible for turbidity. A screening device is also usually required; nevertheless, most fish canning industries already have implemented this kind of pre-treatment.

The results obtained show that sedimentation/flocculation for 1.5 h, was very effective in removing oil and grease and settleable solids. The mean values for TSS and O₂G removals were 48% and 75%, respectively. Since this physical treatment does not promote a significant degradation of organic matter, DOC removal was only 4%. These removal efficiencies are similar to values reported in the literature for other fish processing wastewaters. Muthukumaran and Baskaran (2013) concluded that the majority of the suspended solids that contribute to the organic load would be difficult to remove using a typical sedimentation system.

Following the sedimentation/flocculation stage, a coagulation/flocculation step was employed to remove the resistant suspended and colloidal materials still present in the wastewater. The most commonly used coagulants in wastewater treatment are inorganic compounds (Amuda and Amoo, 2007; de Sena et al., 2008; Wang et al., 2007). So, aluminum sulfate, ferric sulfate, ferric chloride, calcium chloride and polyaluminum chloride were employed in initial experiments at coagulant doses of 100, 200 and 400 mg/L. The results presented in Table 3(a) show that there is not a single optimum coagulant for the chemical treatment of the wastewaters under study. The optimal coagulant and the respective optimal dosage depend on what it is wanted to remove in the moment. However, ferric chloride was found to be an effective coagulant in reducing both TSS (85.8%) and O₂G (99.2%) in the fish canning wastewater, at a dosage of 400 mg/L. Regarding DOC, none of the studied coagulants achieved significant removals. Other works also showed that the coagulant to use depends on the intended purpose. Amuda and Alade (2006) used alum, ferric chloride and ferric compounds.

| Parameter | Min | Max |
|-----------|-----|-----|
| pH        | 6.13| 7.14|
| Conductivity (20 °C) (mS cm⁻¹) | 4.73 | 24.6 |
| TSS (mg L⁻¹) | 324 | 3150 |
| VSS (mg L⁻¹) | 315 | 2680 |
| DOC (mg C L⁻¹) | 90 | 2342 |
| COD (mg O₂ L⁻¹) | 1147 | 8313 |
| BOD₅ (mg O₂ L⁻¹) | 463 | 4569 |
| P_total (mg P L⁻¹) | 13 | 47 |
| N_total soluble (mg N L⁻¹) | 21 | 471 |
| N_ammoniacal (mg NH₄ L⁻¹) | 3 | 1059 |
| Oil and grease (mg L⁻¹) | 156 | 2808 |
| Anions | | |
| F⁻ (mg L⁻¹) | 7 | 60 |
| Cl⁻ (mg L⁻¹) | 174 | 5047 |
| NO₂⁻ (mg N L⁻¹) | 3 | 355 |
| SO₄²⁻ (mg L⁻¹) | <0.01 | 91 |
| Br⁻ (mg L⁻¹) | <0.01 | 214 |
| NO₃⁻ (mg N L⁻¹) | <0.07 | <0.07 |
| PO₄³⁻ (mg P L⁻¹) | <0.00 | 9 |
| Cations | | |
| Li⁺ (mg L⁻¹) | 0 | 1 |
| Na⁺ (mg L⁻¹) | 86 | 2120 |
| NH₄⁺ (mg L⁻¹) | 5 | 217 |
| K⁺ (mg L⁻¹) | 5 | 193 |
| Mg²⁺ (mg L⁻¹) | 7 | 40 |
| Ca²⁺ (mg L⁻¹) | 60 | 221 |
sulfate for abattoir wastewater treatment by coagulation/flocculation and found that alum was more effective in the reduction of TSS, whereas ferric sulfate was more effective in the COD reduction. Braz et al. (2010) also studied winery wastewater treatment by coagulation/flocculation using four different coagulants concluding that the best turbidity removals were achieved with aluminum sulfate, while TSS higher removals were obtained with calcium hydroxide. The same authors also reported the slight ability of coagulation/flocculation to remove COD from winery wastewaters.

Organic coagulants can also be used in coagulation/flocculation. So, two organic coagulants (RIPOL 070 and RIFLOC 1815) in the range 20–400 mg/L were also employed (Cristovão et al., 2012). Nevertheless, the best results obtained with these coagulants (78.7% of TSS and 99.2% of O\textsubscript{G} with 150 mg/L RIFLOC 1815) were not better than those obtained with the inorganic coagulants.

Another parameter that may influence this type of treatment is the pH. Therefore, in addition to the studies carried out at the raw wastewater pH, tests on the influence of pH in the range from 5.0 to 9.0 were also performed, in the best conditions previously encountered. The obtained results (Table 3(b)) show that despite the high removals obtained at alkaline pH, the best removals occurred in the tests carried out without wastewater pH adjustment. Similar results were achieved by Martín et al. (2011) in the coagulation/flocculation treatment of wastewater resulting from sauce manufacturing. Optimization showed similar results under all pH conditions. In this work, alkaline pH was selected as it occurred in the tests carried out without wastewater pH adjustment.

The removals obtained using 400 mg/L of ferric chloride led to a wastewater with suitable characteristics to forward to a secondary biological process for organics removal.

### 4.3. Secondary treatment by activated sludge

The previous steps, sedimentation/flotation and coagulation/flocculation, were not very effective in DOC removal, because soluble organics contribute significantly to the overall organic matter content of the fish canning wastewater. Likewise, the pretreatment wastewater requires removal of organic components by a biological treatment process. In this case an aerobic biological treatment by activated sludge was employed. The biological aerobic reactor operated at HRT of 5, 6 and 8 h. The obtained results are presented in Table 4 and indicate that the biological treatability of this type of wastewaters, under aerobic conditions, is very satisfactory, since low DOC values were achieved at the reactor outlet. The highest DOC removal efficiency (95.8%) corresponds to the highest HRT, which means that, if necessary, higher hydraulic retention times could be used, but greater aeration tanks would be required. The biological treatment of fish processing wastewaters was also studied by other researchers. Riaño et al. (2011) studied the treatment of a fish processing wastewater in two photobioreactors inoculated with microalgae from a lagoon containing aerobically treated swine slurry and with sludge from a membrane submerged bioreactor treating winery wastewater. In such system approximately 70% of total chemical oxygen demand (COD) removal was reached. A new pilot scale hybrid biofilm-suspended biomass membrane bioreactor was used by Artiga et al. (2008) to treat wastewaters generated in a fish canning factory. A COD removal efficiency of 92% was achieved after adaptation of the sludge to the salinity. Comparing these organic matter removal values with that obtained in this work, it is possible to confirm the feasibility and the reliability of the treatment by activated sludge, since the removals obtained were even higher.

### 4.4. Tertiary treatment by filtration, reverse osmosis and UV disinfection

Due to the high chloride concentration and to the presence of microorganisms, a polishing process by reverse osmosis and UV disinfection is necessary in order to reuse fish canning wastewater. The biological treatment by activated sludge significantly reduced the organic content of the wastewater, but suspended biomass still present after clarification may cause fouling of reverse osmosis (RO) membranes. Thus, to reduce the fouling of reverse osmosis membrane and to prolong its life, a tertiary filtration of the pretreated wastewater is required.

So, the biologically treated effluent was passed through a rapid sand filter, before being pumped to reverse osmosis and UV disinfection processes. TSS values before and after the sand filter were measured and an appreciable reduction of TSS was achieved (78%). This removal efficiency lies within the values reported in the

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**Table 3**

| pH | Coagulant | O&G removal (%) | TSS removal (%) | DOC removal (%) |
|----|-----------|-----------------|-----------------|----------------|
|    | Al\textsubscript{2}(SO\textsubscript{4})\textsubscript{3}·16H\textsubscript{2}O | 98.8 | 99.4 | 99.0 |
|    | Fe\textsubscript{2}(SO\textsubscript{4})\textsubscript{3} | 95.4 | 99.5 | 99.2 |
|    | CaCl\textsubscript{2} | 97.0 | 95.6 | 94.7 |
|    | PAC–18 | 99.1 | 99.5 | 98.9 |
|    | FeCl\textsubscript{3} | 98.4 | 99.5 | 99.0 |

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**Table 4**

| HRT (h) | DOC\textsubscript{inlet} (mg/L) | DOC\textsubscript{out} (mg/L) | DOC removal (%) |
|---------|-------------------------------|-------------------------------|----------------|
| 5       | 696                           | 183                           | 73.7           |
| 6       | 620                           | 85                            | 86.2           |
| 8       | 594                           | 25                            | 95.8           |
The aim of this study was to obtain treated water with enough quality to be reused in the industrial plant. The limit of salt content in waters for human consumption, measured as conductivity at 20 °C, is 2500 μS/cm (European Directive 98/83/EC). Since reverse osmosis is an appropriate technology to remove salts from effluents, several tests were conducted in the RO unit (Fig. 4), at permeate flow rates between 192 and 213 L/h, concentrate flow rates between 60 and 270 L/h and pressures between 8.5 and 10.0 bar. Ten trials were performed taking into account these variations as described in Table 5. A complete water analysis was done on permeates from the RO membrane. The DOC, N total soluble and conductivity removals obtained at different operating conditions are presented in Table 5. It seems that for a similar pressure, the greater the permeate flow rate and the lower the concentrate flow rate, the better performance is achieved. The DOC and the conductivity removals increase from 89.6 to 97.3% and from 98.4 to 99.1%, respectively, when increasing the permeate flow rate from 192 to 216 L/h. The best DOC, N total soluble and conductivity removals (97.3, 99.8 and 99.1%, respectively) were achieved for the following conditions (test no. 10): permeate flow rate of 216 L/h, concentrate flow rate of 60 L/h and pressure of 9.0 bar. Under the conditions of test no. 10, the permeate was also analyzed for cations and anions, and the following removals were obtained: 98.5% for sodium, 97.0% for ammonium, 99.2% for potassium, 99.0% for magnesium, 98.5% for calcium, above 96.0% for lithium, 99.2% for chloride, 97.1% for nitrite, 99.8% for sulfate, 96.3% for bromide and 99.2% for phosphate. The overall performance of RO led to negligible concentrations of DOC, N total soluble, conductivity, anions and cations (estimated salinity of 0.03 ppt), as shown in Table 6. Similar results were obtained by other authors. Scholz et al. (2005) showed that a combined treatment of MBR and RO allowed for 90–100% reduction of COD, BOD and ammonia in a mixed-tannery effluent. Salt content was reduced by 97.1%. Jin et al. (2013) verified that the combination of biological and membrane processes, including reverse osmosis, applied to coking wastewater led to water suitable for industrial reuse. Reuse of tannery wastewater was investigated by Ranganathan and Kabadgi (2011) employing conventional treatment methods like neutralization, clari-flocculation and biological processes as pre-treatment before RO separation process. About 93–98%, 92–99% and 91–96% removal of TDS, sodium and chloride, respectively, were achieved in 5 different tanneries. A large amount (70–85%) of water could be recovered and recycled.

Considering these results, it may be concluded that the present solution has substantial potential in the treatment of fish canning industrial wastewater for subsequent reuse as process water or for irrigation. However, it is also necessary to ensure that the microbiological quality meets the legislation for water reuse in food industry (European Directive 98/83/EC – Table 1), which requires the application of UV disinfection to the permeate of RO process.

So, when treating the effluent from RO, performed at the best conditions, by UV radiation (Fig. 5), a count of heterotrophic bacteria was carried out both before and after the disinfection process, in order to assess the treatment effectiveness for the intended purpose. After incubation at 22 and 37 °C for 24 and 48 h, it was found that the samples prior to UV disinfection still have plenty of microorganisms: 140,000 CFU/100 mL at 22 °C and 10,000 CFU/100 mL at 37 °C. However, when the count was performed after disinfection, it was found that UV radiation is really effective in the effluent disinfection, as 100% removal of the bacteria present in the feed stream was achieved.

UV disinfection has also proved to be effective as regards other types of effluents to be reused in the industrial process. Mavrov and Belières (2000) studied the treatment of three low-contaminated process waters from the food industry by a combination of pre-treatment, membrane filtration and UV disinfection, achieving a treated water with quality enough to be reused in accordance with company’s needs. Poultry slaughterhouse wastewater reclamation was investigated by de Nardi et al. (2011), using a lab-scale treatment system consisting of biological degradation (sequencing bath reactor), dissolved air flotation and UV disinfection. The final effluent met the quality standards of the legislation for both potable water and effluents to be discharged into receiving water bodies.

Taking into account the characteristics of the clarified effluent after the overall treatment proposed (Table 6) and knowing that UV disinfection is 100% effective, one can conclude that the treated water meets the values established for water intended for human consumption (European Directive 98/83/CE). The water reuse in the manufacturing process or, alternatively, for washing floors, irrigation, etc. is also possible. The accumulated removal efficiencies in the different sequential treatment stages are presented in Fig. 6. Removals of 99.9% for DOC, 99.8% for O₂G, 98.4% for TSS, 99.1% for conductivity, above 96% for anions and cations and 100% for heterotrophic bacteria (CFU) were attained at the end of the proposed treatment sequence.

### 4.5. Energy and chemicals costs

Data on energy and chemicals costs to operate the wastewater treatment process at a flow rate of 10 m³ per day (working time 8 h day⁻¹) are summarized in this section. Operating costs regarding salaries, maintenance and others were not considered as they are less significant.

| Test | Permeate flow rate (L/h) | Concentrated flow rate (L/h) | Pressure (bar) | DOC (%) | N total soluble (%) | Conductivity (%) |
|------|-------------------------|-----------------------------|---------------|---------|-------------------|-----------------|
| 1    | 192                     | 270                         | 8.5           | 93.9    | 95.2              | 98.7            |
| 2    | 192                     | 240                         | 8.5           | 96.6    | 96.0              | 96.0            |
| 3    | 192                     | 204                         | 9.0           | 89.6    | 95.0              | 98.4            |
| 4    | 198                     | 90                          | 9.0           | 96.0    | 95.9              | 98.4            |
| 5    | 198                     | 120                         | 9.3           | 96.9    | 96.1              | 98.4            |
| 6    | 198                     | 180                         | 10.0          | 96.1    | 96.2              | 98.1            |
| 7    | 198                     | 225                         | 9.0           | 90.1    | 99.8              | 95.7            |
| 8    | 210                     | 120                         | 9.0           | 95.5    | 98.8              | 99.9            |
| 9    | 210                     | 240                         | 8.5           | 93.9    | 99.8              | 99.0            |
| 10   | 216                     | 60                          | 9.0           | 97.3    | 99.8              | 99.1            |
The first process comprises a flotation unit, 5 m³/h maximum flow capacity, 2.7 kW electric power (including energy consumption for surface scraper). So, the energy consumption is 5.4 kWh per day.

Coagulation/flocculation unit comprises two steps: rapid mixing in a tank of 200 L during 5 min provided by an electrical agitator (shaft, coupling and four-bladed propeller at 45°, in stainless steel), a 3-phase current motor 230/400 V, nominal power of 0.55 kW and nominal speed 106 rpm (2.31 kWh per day); slow mixing at 32 rpm in a 1000 L tank during 30 min also using an electrical agitator (shaft, coupling and propeller type 2R, in stainless steel), a 3-phase current motor 230/400 V with nominal power of 0.37 kW (energy consumption of 4.16 kWh/d). In this process the optimum results were obtained with a FeCl₃ dosage of 400 mg/L, which corresponds to a daily consumption of 4 kg. At a local supplier the cost of this chemical is about 290 €/tonne.

One of the main stages in the treatment plant is the activated sludge process. The mixture and aeration in the biological reactor (total volume of 10 m³) is achieved by a submersible aerator (0.55 kW) operating 10 h a day and, in a 5 m² secondary clarifier (4 h detention time), a pump is installed for sludge recirculation (nominal power of 1.4 kW, 2 h per day). This contributes to a daily energy consumption of 8.3 kWh.

Before RO the residual suspended solids are removed in a pressurized multimedia filter unit equipped with an automatic valve: flow rate of 1400 L/h (at maximum filtration velocity 15 m/h), 0.09 m² filtration area and working pressure 4–6 bar. According to the manufacturer this requires a 3000 L/h backwash flow (1.5–2.0 times the flow rate). So, a feeding/recirculating pump of 0.75 kW nominal power is needed to ensure $Q_{\text{max}} = 3 \text{ m}^3/\text{h}$ at $H_{\text{max}} = 3 \text{ m.c.a.}$ during a 20 min backwash daily period.

A reverse osmosis plant designed for a maximum flow rate of 250 L/h (permeate flow rate equal to 75% of the inlet flow), including three RO membranes 40 × 40 (1016 × 180 × 180 mm) and a multistage pump of 2.2 kW/400V motor can achieve the defined goals working 15 h a day.

Finally, the permeate (7500 L/day) was subjected to ultraviolet radiation in a cylindrical stainless steel reactor to promote its disinfection, using a 23 W low-pressure lamp operating continuously.

Table 7 summarizes the energy and the amount of coagulant (FeCl₃) consumed in each stage of the wastewater treatment process, as well as the total cost associated. The value obtained considers the unit costs of energy and ferric chloride, 0.1416 €/kWh and 290 €/tonne, respectively. On this basis, the unit treatment cost to obtain a treated effluent suitable for reuse in the process is 0.85 € per cubic meter. As shown in Table 7, we can reduce costs by about 60% if the goal is only to treat the wastewater to meet the legislated discharge limits, i.e., if the tertiary treatment is excluded. Taking into account that the tap water cost for the manufacturing process, including taxes, is approximately 2.1 € per cubic meter, it is possible to conclude that even considering personnel, maintenance and produced sludge treatment/disposal costs, the wastewater reuse, after treatment, will be a benefit to the fish canning industry in study.

5. Conclusions

The fish canning industry wastewater treatment was investigated in order to obtain water with quality requirements to reuse in the industrial process.

The proposed sequential system, combining sedimentation/flocculation, coagulation/flocculation, aerobic biological degradation by activated sludge, filtration, reverse osmosis and UV disinfection, proved to be very effective. Removals of 48% for TSS and 75% for O₆G were observed in the sedimentation stage. For coagulation/flocculation, ferric chloride showed to be the better coagulant at a dosage of 400 mg/L and raw wastewater pH. O₆G and TSS removals attained 99.2 and 85.8%, respectively. DOC was reduced by 95.8% in the biological process by activated sludge for HRT = 8 h. In the tertiary treatment, the rapid sand filtration retained 78% of TSS whereas reverse osmosis removed 97.3% of DOC, 99.8% of $N_{\text{total}}$ soluble, 99.1% of conductivity/salinity and above 96% of the analyzed ions. The removal of heterotrophic bacteria (CFU) by UV disinfection reached 100%.

The overall removals attained by the treatment sequence under study were: 99.9% for DOC, 99.8% for O₆G, 98.4 for TSS, 99.1% for conductivity/salinity, above 96% for the analyzed ions and 100% for heterotrophic bacteria, which led to achieve, as intended, water with quality requirements for reuse in the industrial unit.

The treatment cost is 0.85 € per cubic meter but it can be reduced by about 60% if the goal is only to treat the wastewater in order to meet the legislated discharge limits.

The implementation of this reuse process at large scale will bring both environmental and economic benefits, since it allows the reduction of the effluent to be discharged and the water consumption, thus decreasing the associated costs.

![Fig. 6. Accumulated removal efficiencies of the different sequential stages of the fish canning wastewater treatment.](image-url)
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