First demonstration of an all-solid-state optical cryocooler

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Abstract
Solid-state optical refrigeration uses anti-Stokes fluorescence to cool macroscopic objects to cryogenic temperatures without vibrations. Crystals such as Yb3+-doped YLiF4 (YLF:Yb) have previously been laser-cooled to 91 K. In this study, we show for the first time laser cooling of a payload connected to a cooling crystal. A YLF:Yb crystal was placed inside a Herriott cell and pumped with a 1020-nm laser (47 W) to cool a HgCdTe sensor that is part of a working Fourier Transform Infrared (FTIR) spectrometer to 135 K. This first demonstration of an all-solid-state optical cryocooler was enabled by careful control of the various desired and undesired heat flows. Fluorescence heating of the payload was minimized by using a single-kink YLF thermal link between the YLF:Yb cooling crystal and the copper coldfinger that held the HgCdTe sensor. The adhesive-free bond between YLF and YLF:Yb showed excellent thermal reliability. This laser-cooled assembly was then supported by silica aerogel cylinders inside a vacuum clamshell to minimize undesired conductive and radiative heat loads from the warm surroundings. Our structure can serve as a baseline for future optical cryocooler devices.

Introduction
All-optical cooling of a solid was first observed in 1995 by Epstein et al.1, and extensive developments over the past two decades in materials, characterization techniques, and optical designs have laid the groundwork for practical applications. Early work focused on the cooling of Yb3+-doped fluoride glasses (e.g., ZBLANP:Yb3+), the cooling efficiency of which was later found to be limited by the substantial inhomogeneous broadening of the Yb3+-fold transition in the amorphous glass host2. The much smaller inhomogeneous broadening in Yb3+-doped fluoride crystals (e.g., YLiF4:Yb3+) allowed for higher cooling efficiencies, which helped enable the breakthrough into the cryogenic regime in 20103. This breakthrough has fueled further research into solid-state optical refrigeration4,5, which is currently the only technology that can provide truly vibration-free cooling to cryogenic temperatures6.

Solid-state laser cooling is achieved using anti-Stokes fluorescence, a process in which the average wavelength of the fluorescence (λf) emitted by a material is shorter than the wavelength (λ) of the laser used for excitation7. The cooling efficiency ηc is given by the ratio of the cooling power (Pcool) and absorbed power (Pabs):8

\[
\eta_c = \frac{P_{\text{cool}}}{P_{\text{abs}}} = \frac{\eta_{\text{ext}} \eta_{\text{abs}} \lambda}{\lambda_f} - 1
\]

where ηext = ηeWf/(ηeWf + Wnr) is the external quantum efficiency and ηabs = αr(λ)/(αr(λ) + αb(λ)) is the absorption efficiency. Here, Wf and Wnr are the radiative and non-radiative decay rates of the emitting state, respectively; ηe is the fluorescence escape efficiency; αr(λ) is the resonant absorption coefficient of the cooling transition; and αb(λ) is the parasitic background absorption coefficient. The radiative process of cooling by anti-Stokes fluorescence competes with a variety of non-radiative processes that reduce ηext and ηabs by multiphonon relaxation of the...
emitting state ($W_{nr}$) and absorption by impurities ($a_{ib}$), respectively. Yb$^{3+}$-doped solids have received particular attention because Yb$^{3+}$ can be excited with high-power lasers of ~1 μm wavelength and can emit with high quantum efficiency due to its simple energy-level structure. However, laser cooling of Yb$^{3+}$-doped solids to cryogenic temperatures is only practical if $\eta_{ext}\eta_{abs}$ exceeds 0.98–0.99, a challenging requirement that necessitates a cooling crystal with exquisite purity. High-purity YLiF$_4$ crystals doped with 10 mol% Yb$^{3+}$ (YLF:10%Yb) have recently been cooled to 91 K by exciting them with a laser tuned to the E4→E5 crystal-field transition of Yb$^{3+}$ at 1020 nm.

Most solid-state laser cooling studies to date have focused on the optical refrigeration of the laser-cooling material itself or a load that is transparent to fluorescence. For practical applications, however, it is necessary for the laser-cooling material to refrigerate an attached arbitrary external payload, such as a sensor or electronic component. The primary challenges for advancing from a basic laser-cooling setup to a practical optical cryocooler device involve: (1) managing the numerous heat and radiation flows in the system and (2) providing a sturdy, thermally insulating support structure for the laser-cooled assembly. In this article, we address these challenges and describe the design and first experimental demonstration of a solid-state optical cryocooler capable of refrigerating a HgCdTe infrared sensor to 135 K. This device is the only solid-state cooling device that works in the cryogenic regime, i.e., in the temperature range of liquid cryogens, such as Xe, with a boiling point of 166 K. The most common solid-state refrigerators, thermoelectric coolers, cannot achieve temperatures nearly as low as 135 K.

Figure 1 shows a block diagram of the solid-state optical cryocooler architecture used in this study. The payload is attached via a coldfinger, mirror, and thermal link to the laser-cooling material. This assembly (refrigerator cold side) is mounted with a support element and surrounded by a closely fitting clamshell (refrigerator hot side), which is lined with a low-emissivity coating. The breakthrough performance of the present solid-state optical cryocooler is enabled by the materials and geometries of both the thermal link and support element combined with a high-performance YLF:10%Yb$^{3+}$ crystal. The design of various components and the performance of the optical cryocooler are discussed in detail in the following sections.

Materials and methods

Optical components

We used the YLF:10%Yb laser-cooling crystal that was cooled to 91 K in a previous study. This crystal was cut so that the linearly polarized pump laser has $\mathbf{E} \parallel \mathbf{\hat{c}}$ because the absorption cross section of the E4→E5 crystal-field transition.
transition in a YLF:Yb crystal at 1020 nm is $\approx 1.7 \times$ greater in this geometry than that with $E \perp \vec{c}$ polarization\textsuperscript{12}.

The thermal link was fabricated from a Czochralski-grown undoped YLF crystal boule (AC Materials, Tarpon Springs, FL), which was found to have low parasitic background absorption coefficients of $a_b(\vec{E} \parallel \vec{c}) = 9.6 \times 10^{-5}$ cm$^{-1}$ and $a_b(\vec{E} \perp \vec{c}) = 4.4 \times 10^{-5}$ cm$^{-1}$, as determined by exposing the sample to a 1020-nm laser (43 W) and measuring the resulting temperature increase relative to a reference heat load using a thermal camera (Nanocore 640, L3 Communications Corporation, Garland, TX, USA). The YLF thermal link was fabricated such that its $\vec{c}$ axis was aligned parallel to the $\vec{c}$ axis of the YLF:10%Yb crystal at the respective mating interface (Fig. 1). This crystallographic alignment minimized the thermal stresses induced by the anisotropic coefficients of thermal expansion (CTE) of YLF. The thermal link surface that was attached to the coldfinger was subsequently coated with a metallic mirror by first etching the YLF in an argon plasma followed by sputter deposition of 200 nm of silver, followed by 200 nm of gold. This produced a silver mirror with high reflectance on the inside of the thermal link that was protected from oxidation by the gold layer. Figure 2a, b show the mirrored YLF thermal link before and after Adhesive-Free Bonding\textsuperscript{13,14} (AFB\textsuperscript{®} by Onyx Optics, Inc., Dublin, CA) to the YLF:10%Yb crystal, respectively. The elevated temperatures during the AFB process caused interdiffusion of the Ag and Au layers, as shown by the discoloration of the external gold layer in Figure 2b. No discoloration of the optically important inside Ag mirror was observed. Furthermore, no mechanical failure was observed when temperature cycling an equivalent assembly 22 times from 300 to 75 K (at $-6.9$ K/min) and back to 300 K (at $+21.6$ K/min), demonstrating that a thermo-mechanically reliable AFB between undoped YLF and YLF:10%Yb was achieved.

Silica aerogels were synthesized by a sol–gel process, followed by drying in supercritical methanol, which yielded hydrophobic aerogels that reversibly adsorbed $<2$ wt% water during long-term storage in ambient air\textsuperscript{15}. This process also allowed custom aerogel shapes to be directly molded during synthesis. We fabricated hydrophobic silica aerogels with a 0.1 g/cm$^3$ density in the shape of 5 mm diameter $\times$ 5 mm tall cylinders as the basic support element. The samples were found to absorb $<2 \times 10^{-4}$ of the incident power of a 1015-nm laser and were therefore expected to produce a negligible heat load when exposed to YLF:Yb fluorescence.

The solid-state optical cryocooler was placed in a vacuum chamber ($5 \times 10^{-7}$ Torr), and the YLF:10%Yb crystal was excited by a linearly-polarized continuous-wave fiber laser (IPG Photonics, Inc., custom-made) that provided 47 W at 1020 nm. The temperature of the coldfinger was monitored with a calibrated silicon diode (LakeShore Cryotronics, DT-670-SD) located in the coldfinger base. In addition, non-contact differential luminescence thermometry (DLT)\textsuperscript{3} was used to monitor the temperature of the cooling crystal by collecting fluorescence with a multimode optical fiber inserted flush with the clamshell wall near the YLF:10%Yb crystal.

**Thermal model**

The dynamics of the observed temperature changes of the cooling crystal and coldfinger follow an intricate interplay between the absorbed laser power, cooling efficiency, parasitic losses (due to fluorescence absorption) at interfaces, radiative loads, conductive loads, and useful loads (e.g., heat lift from the sensor). A rigorous model that numerically solves the 3-dimensional heat equation including the exact structure of the optical cooler could, in principle, reproduce the observed thermal dynamics. However, we were able to determine the essential features and underlying parameters by considering a simple 3-element model, as depicted in Figure 3, that consisted of the YLF:Yb cooling crystal (at temperature $T_x$) connected to the coldfinger (at temperature $T_f$) by a thermal link (at

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**Fig. 2 Geometry, fabrication, and bonding of the YLF thermal link.** a Fabricated YLF thermal link with a sputtered Ag/Au mirror on the coldfinger interface b YLF thermal link AFB-bonded to the Brewster-cut YLF:10%Yb cooling crystal. c Cross sectional view and dimensions (in mm) of the $\delta = 15^\circ$ geometry with a $90^\circ$ inside angle. The dimension in the orthogonal direction is 4.10 mm. The orientation of the $\vec{c}$ axis is indicated, and it is parallel to the $\vec{c}$ axis of the YLF:10%Yb crystal.
temperature $T_l$), which in this case, was also constructed from a YLF (undoped) crystal.

Assuming that the adhesive-free bond between the cooling crystal and thermal link introduces negligible thermal resistance and parasitic loss, the equations governing the temperature evolution of the three elements are approximated as follows:

\[
\frac{dT_x}{dt} = -\eta_s(T_x)P_{abs}(T_x) + \epsilon_s\sigma(T_x - T_c)A_x + K_l(T_x)(T_l - T_x)
\]

\[
\frac{dT_l}{dt} = -K_l(T_x)(T_l - T_x) + \epsilon_l\sigma(T_l^4 - T_l^4)A_l + K_f(T_l)(T_f - T_l)
\]

\[
\frac{dT_f}{dt} = -K_f(T_l)(T_f - T_l) + \epsilon_f\sigma(T_f^4 - T_f^4)A_f + \beta_1\eta_{ext}(T_x)P_{abs}(T_x) + P_{lift}
\]

where $C_{x,f,l} = V_{x,f,l}(T)$, with $V_{x,f,l}$ denoting the known volume and heat capacity (J/K/m$^3$), respectively, of the cooling crystal (x), link (l), and coldfinger (f).

The temperature dependence of the YLF heat capacity is taken from Ref. 16. $P_{abs}(T) = P_{in}\eta_{cpl}$ is the absorbed power drawn from the incident power $P_{in}$ (at $\lambda = 1020$ nm). Here, $\eta_{cpl} = 1 - \exp(-2N_{rt}\alpha_s(\lambda,T)L_x)$ is the laser coupling efficiency, where $N_{rt}$ is the number of roundtrips the laser makes through the crystal of length $L_x$ placed inside the astigmatic Herriott cell\(^\text{17}\) and $\alpha_s(\lambda,T)$ is the resonant absorption coefficient of the Yb$^{3+}$ crystal-field transition\(^\text{17}\). $K_f(T_l) = 2k_A L_f$ is the effective thermal conductance of the link with a cross-sectional area of A and an effective length $L_f$. The factor of 2 arises from the fact that $T_l$ is the temperature of the link halfway between the crystal and the coldfinger. $\chi(T_l)$ is the temperature-dependent thermal conductivity of the thermal link (undoped YLF)\(^\text{16}\). $A_s$, $A_p$, and $A_f$ denote the total surface area of the cooling crystal, link, and coldfinger, respectively. $\epsilon_s \approx \epsilon_l$ are the effective emissivities, $\epsilon_l/(1 + \chi)$, of YLF in a tight clamshell coated with a low emissivity ($\epsilon_s$) TINOX solar absorber (Almeco GmbH, Bernburg, Germany) held at constant temperature $T_c$ and $\epsilon_l$ is the effective emissivity of the copper coldfinger. Here, $\chi = (1-\epsilon_s)\epsilon_l A_l/\epsilon_c A_c$ (Ref. [11]), where $\epsilon_l$ and $\epsilon_s$ are the emissivities and $\sigma$ is the Stefan-Boltzmann constant. $\beta_1$ represents the fraction of the total fluorescence power that effectively leaks through the link and is consequently absorbed by the coldfinger. $P_{lift}$ is the useful thermal load (payload heat lift) and small parasitic load from the sensor wires and aerogel supports.

**Results and discussion**

**Cooling crystal and Herriott cell**

As shown in Eq. (1), the cooling power of a solid-state optical refrigerator linearly scales with the laser power absorbed by the cooling crystal (YLF:Yb), $P_{abs} = P_{in}\eta_{cpl}$, which in turn depends on $N_{rt}$, $\alpha_s$, and $L_x$. The temperature dependence of $\alpha_s$ scales as $\alpha_s(T_x)\sim(1+\exp(\delta E_f/k_B T_x))^{-1}$, where $\delta E_f = 460$ cm$^{-1}$ is the width of the $^2F_{7/2}$ ground-state multiplet in YLF:Yb, and $k_B$ is the Boltzmann constant\(^\text{17}\). The decrease of $\alpha_s$ at low temperatures requires $N_{rt}$ to be $\sim 13\times$ greater at 135 K than at 300 K to realize the same $P_{abs}$. To realize a large $P_{abs}$, the Brewster-cut YLF:10%Yb crystal was placed inside an astigmatic Herriott cell\(^\text{17}\) formed by a concave spherical input mirror ($R = 50$ cm) and a cylindrical back reflector ($R_s = \infty$, $R_f = 50$ cm) with the laser focused into the cell with a spherical lens ($f = 5$ cm) through a 400 $\mu$m diameter center hole in the input mirror. We estimate $N_{rt} = 40$ in the final configuration, which yields $\eta_{cpl} > 0.9999$ with $L_x = 1.11$ cm and $\alpha_s = 0.12$ cm$^{-1}$ at 1020 nm and 135 K for $\bar{E} \parallel \bar{c}$.

**Thermal link**

Attaching the payload directly to the YLF:Yb crystal is impractical. Although this would provide an excellent thermal contact, the payload would be directly exposed to YLF:Yb fluorescence and absorb radiation that could exceed $P_{cool}$. Therefore, a transparent thermal link was inserted between the YLF:Yb crystal and payload (Fig. 1) to reduce the amount of fluorescence reaching the payload while maintaining an efficient thermal path between the payload and YLF:Yb crystal\(^\text{18,19}\). The thermal, mechanical, and optical properties of the resulting interface between the YLF:Yb crystal and thermal link require careful consideration. The use of adhesives to bond the two elements can introduce optical absorption and corresponding heating. We avoided this by bonding the cooling crystal and thermal link by van-der-Waals forces in an Adhesive-Free Bond (AFB)\(^\text{13,14}\). Although sapphire is an attractive thermal-link material due to its low optical...
absorption and high thermal conductivity (κ ≈ 400 W/m K at 100 K\(^{-1}\)), our preliminary experiments exploring the AFB of YLF:10%Yb and sapphire found frequent bond failures upon thermal cycling from 300 to 77 K. The stress at the AFB interface induced by a temperature change \(\Delta T\) is proportional to \(\Delta T \Delta \alpha\), where \(\Delta \alpha\) is the difference in the coefficients of thermal expansion (CTE) of the two bonded materials\(^{21}\). The CTE of sapphire (\(\alpha_s = 5.22 \times 10^{-6} K^{-1}\) and \(\alpha_p = 5.92 \times 10^{-6} K^{-1}\) at 296 K\(^{20}\)) and YLF (\(\alpha_s = 13.0 \ldots 14.3 \times 10^{-6} K^{-1}\) and \(\alpha_c = 8.0 \ldots 10.1 \times 10^{-6}\) at 300 K\(^{17,22}\)) indicate a \(\Delta \alpha = (8 \ldots 9) \times 10^{-6} K^{-1}\) and a resulting stress of 180–200 MPa for \(\Delta T = 223 K\), which exceeds the tolerable stress of mechanically reliable AFB composites of dissimilar materials\(^{21}\) by \(\sim 10\times\). We therefore chose to use undoped YLF instead of sapphire as the thermal link material, nominally yielding \(\Delta \alpha = 0\) and thus a stress-free adhesive-free bond between the YLF:10%Yb cooling crystal and the undoped YLF thermal link. However, this results in a \(\sim 14\times\) lower thermal conductivity of YLF (\(\kappa_s = 24 W/m K\) and \(\kappa_c = 34 W/m K\) at 100 K\(^{16}\)) than that of sapphire. The undoped YLF thermal link, therefore, must be as short as possible to minimize the temperature difference between the payload and YLF:Yb crystal, while still providing good fluorescence light rejection.

Given these constraints, we optimized the YLF thermal link in the geometry of a short kinked YLF waveguide (Fig. 4a, inset). The goal was to minimize the total heat load introduced by the thermal link \(P_t = P_{\text{fluor}} + P_{\text{rad}}\), where \(P_{\text{rad}}\) is the fraction of fluorescence power absorbed at the thermal link mirror surface (see Fig. 1) and \(P_{\text{rad}}\) is the heat load due to radiative heat transport from the coated clamshell surface (warm) to the thermal link (cold). Our measurements have shown that negligible heat is produced by the absorption of the fluorescence in the bulk of the YLF thermal link. We performed an optical raytracing simulation of the light transport in the thermal link using the FRED Optical Engineering software (Photon Engineering, LLC, Tucson, AZ, USA). A \(4.1 \times 4.1 \text{ mm}^2\) link cross section was chosen to match the width of the existing YLF:10%Yb crystal. The model used \(\lambda_f = 1004.7\text{ nm}\) as measured for YLF:Yb at 135 K, and a corresponding averaged refractive index of \(n = (2n_0 + n_e)/3 = 1.456\), where \(n_0 = 1.448\) and \(n_e = 1.471\)\(^{23}\), was used for both undoped YLF and YLF:10%Yb. The fluorescence was represented by \(10^6\) rays that were uniformly and isotropically emitted within the attached YLF:Yb crystal. As shown in Figure 4a, an exponential decrease of the fraction of fluorescence power reaching the link/mirror
interface $\eta_L$ occurs when increasing the tilt angle $\delta$; $\eta_L$ does not significantly change for $\delta > 15^\circ$. This simulation yields $P_{\text{fluor}} = P_{\text{in}}\eta_{\text{cpl}}\eta_{\text{ext}}\eta_L(1 - R_m)$, where $R_m$ is the reflectance of the mirror, which is set as 0.97 for the sputtered silver layer used in this study, and $P_{\text{in}} = 47$ W, $\eta_{\text{cpl}} = 0.999$, and $\eta_{\text{ext}} = 0.996$ are assumed. The radiative heat load on the thermal link is approximately $P_{\text{rad}} = \varepsilon_l A_l \sigma (T_l^4 - T_c^4) / (1 + \chi)$. A 1-mm gap is assumed between the thermal link and the TiNOX-coated clamshell surfaces as well as $\varepsilon_c = 0.9$ (from CaF$_2^{24}$), $\varepsilon_c = 0.05$ (TiNOX solar absorber datasheet), $T_l = 135$ K, and $T_c = 300$ K. Figure 4b plots $P_{\text{rad}}$ vs. $P_{\text{fluor}}$ for thermal links with different $\delta$. The $\delta = 15^\circ$ design achieves the lowest $P_{\text{L}}$, whereas links with a smaller $\delta$ have a significantly larger $P_{\text{fluor}}$ due to the greater $\eta_L$, and links with greater $\delta$ have a slightly larger $P_{\text{rad}}$ due to their larger surface areas $A_l$ and $A_c$. The link performance is slightly improved by providing a 90° inside angle (see inset Fig. 4a), a geometry that is also more favorable for fabrication than geometries with smaller inside angles. The final thermal link design (Fig. 2c) is estimated to produce a total heat load of $P_L = 31.3$ mW, of which 75% is due to residual mirror absorption. Compared with a typical $P_{\text{cool}} = 470$ mW (for $P_{\text{in}} = 47$ W, $\eta_{\text{cpl}} = 1$, and $\eta_c = 0.01$), the heat load introduced by the YLF thermal link is relatively small.

**Silica aerogel supports**

The laser-cooled assembly consisting of the YLF:Yb crystal, YLF thermal link, mirror, coldfinger, and payload (Fig. 1) must be mounted within the closely fitting clamshell structure in a manner that introduces minimal thermal conduction from the clamshell to the cooled assembly, is mechanically reliable, and can be easily assembled. Earlier experiments using silica optical fibers with sharpened tips as support elements were difficult to assemble and mechanically unreliable. The new approach implemented in this study uses several silica aerogel cylinders to support the cooled assembly under the coldfinger. silica aerogels are open-celled mesoporous SiO$_2$ networks with low mass densities of typically 0.01–0.2 g/cm$^3$ as well as low optical absorption in the visible and near-infrared spectral range. Pure SiO$_2$
Aerogels have low thermal conductivities of \(~0.004\) W/m K (at 300 K) and \(~0.001\) W/m K (at 100 K) in a vacuum\(^2\), making them attractive for use as support structures in optical cryocoolers. The laser-cooled assembly (19.1 g total weight) rested on three 5 mm diameter \times 5 mm tall aerogel cylinders (58.9 mm\(^2\) total area), thus producing a pressure of 3.18 kPa. Four additional aerogel cylinders were used around the outside perimeter of the cold finger base to laterally secure the assembly. No mechanical degradation of the aerogel cylinders was observed under these conditions, which is consistent with the compressive strength of silica aerogels, for which values of 1 MPa (0.25 g/cm\(^3\))\(^2\) and 0.047–0.11 MPa (0.1 g/cm\(^3\))\(^2\) have been reported.

**Laser-cooled assembly and clamshell**

Earlier solid-state optical cryocooler designs envisioned attaching the payload directly to the coated side of the thermal link\(^1\),\(^9\),\(^2\) an approach that creates significant size constraints for the placement of the support elements and potentially larger payloads. We therefore introduced a copper coldfinger in the present cryocooler, allowing for more space to accommodate larger payloads, aerogel supports, as well as other instrumentation, such as a temperature sensor and heating element. Figure 5a, b show the cooled assembly consisting of the YLF:Yb crystal/YLF thermal link unit and the HgCdTe infrared photo-sensor chip (InfraRed Associates, Inc.) attached to the copper coldfinger. The necked shape of the coldfinger near the thermal link interface was created to provide an optical baffle that prevented residual fluorescence from reaching the payload.

The clamshell serves as the heat sink for the fluorescence emitted by the YLF:Yb crystal and must be designed to minimize the radiative heat loads [Eqs. (2)–(4)] it imparts on the cooled assembly, which requires a geometry that closely envelops the cooled assembly to minimize the surface area and maximize \(\chi\). Figure 5c shows the assembly sequence of the layered structure, which allows for a stepwise installation of the clamshell at a separation of 1 mm around the cooled assembly. We used a TiNOX Energy solar absorber (Almeco GmbH, Bernburg, Germany) glued to the inside surfaces of the clamshell with silver epoxy to provide high absorbance of the YLF:Yb fluorescence and low thermal emissivity. The clamshell also included a viewport that provided optical access to the HgCdTe sensor payload.

**Laser cooling of a HgCdTe sensor**

Figure 6a shows the coldfinger temperature (measured with a silicon diode) and the cooling crystal temperature (measured by DLT) as a function of time after turning on the 1020-nm laser. The 1020 nm laser power was increased in steps from 0, 12, 25, to 47 W as a precautionary measure to reduce the potentially large temperature gradient and associated thermally induced stresses between the YLF:Yb crystal and coldfinger that develop at early times. This is evident in the stepwise decrease of the YLF:Yb crystal temperature as measured by DLT. The coldfinger temperature reached 134.9 K after 4 h.
representing the first ever demonstration of cooling a payload to the cryogenic regime using solid-state optical refrigeration. Furthermore, these data experimentally validate the various geometry and material choices for the thermal link, coldfinger, aerogel supports, and clamshell discussed above.

The results shown in Figure 6a were obtained without an electrical current flowing through the HgCdTe sensor, i.e., without an external heat load introduced by the payload. A separate resistor installed in the coldfinger base allowed for the application of well-defined heat loads and measurement of the corresponding increase in the coldfinger temperature. Figure 6b presents the heat load curve obtained with the YLF:10%Yb crystal pumped at 47 W. The initial slope is 0.31 K/mW, a value that is characteristic for this particular cryocooler. Powering up the HgCdTe sensor through a Fourier-Transform Infrared (FTIR) spectrometer (Midac M4401) at the 134.9 K base temperature resulted in a temperature increase of \( \Delta T = 6.8 \text{ K} \) across the link when the number of roundtrips is increased to \( N_{\text{rt}} = 30 \), as expected in the Herriott cell arrangement. Considering the simplicity of our model, these calculated values are in excellent agreement with previously measured values for this 10% Yb\(^{3+}\)-doped YLF crystal. \( \beta_i \) also agrees with the calculated value from the raytracing model. \( T_e \) was fixed at 283.15 K, which is the final measured temperature of the clamshell. With the above parameters fixed, the model predicts a final coldfinger temperature \( T_f = 135.0 \text{ K} \) with a temperature drop \( \Delta T = T_f - T_e = 6.8 \text{ K} \) across the link when the number of roundtrips is increased to \( N_{\text{rt}} \geq 30 \), as expected in the Herriott cell arrangement. Considering the simplicity of our model, these calculated values are in excellent agreement with our observed values of \( T_f = 134.9 \text{ K} \) and \( \Delta T = 6.1 \text{ K} \) (Fig. 6a). The above calculations assumed a negligible heat lift (\( P_{\text{lift}} = 0 \text{ mW} \)) in comparison to the existing radiative and fluorescence loads. Furthermore, the measured initial slope of 0.31 K/mW (Fig. 6b) is in good agreement with \( \partial T_f/\partial P_{\text{lift}} \approx 0.31 \text{ K/mW} \) obtained from the model calculations.
FTIR spectroscopy with the laser-cooled HgCdTe sensor

In a final experiment, the laser-cooled HgCdTe sensor was used as part of an FTIR spectrometer and the infrared absorption spectrum of a sheet of low-density poly-ethylene (LDPE) was measured (Fig. 8). This is the first demonstration of a sensor cooled by solid-state optical refrigeration used as part of a practical device. Figure 8 shows the absorbance spectrum of the same sample measured with an equivalent HgCdTe sensor cooled with liquid nitrogen to 77 K. The bandgap energy $E_g = \frac{hc}{\lambda}$ of this HgCdTe sensor corresponds to a cutoff wavelength of $\lambda = 16.7 \, \mu m$. The primary noise source is due to statistical fluctuations of the thermally activated dark current within the detector element. The dark current $I_d$ is proportional to the number of carriers that are thermally excited across the bandgap and thus varies approximately as $I_d \propto \exp \left( -\frac{E_g}{k_B T} \right)$, with the corresponding noise amplitude varying as $\sqrt{I_d}$. The detector noise amplitude is therefore expected to be $\sim 11 \times$ greater at 135 K than at 77 K, which is consistent with the observed lower signal-to-noise ratio in the measurement using the laser-cooled sensor.

Conclusions

We used solid-state optical refrigeration to cool a payload to cryogenic temperatures for the first time, which represents a breakthrough in this field and opens the door to using this technology for a variety of applications that benefit from a reliable cryogenic refrigerator without moving parts and associated vibrations. The all-solid-state optical cryocooler was enabled by four key elements: First, a high-quality YLF:Yb crystal with a low $\alpha_b$ and high $\eta_{ext}$ provided efficient cooling at the heart of the cryocooler. Second, the use of a coldfinger provided the design flexibility required to incorporate the sensor payload, support structure, and diagnostic components. Third, the results show that hydrophobic silica aerogels can provide excellent thermal insulation as well as sufficient mechanical strength to support the laser-cooled assembly. Fourth, we demonstrated that a thermo-mechanically reliable van-der-Waals bond can be created between Yb$^{3+}$-doped and undoped YLF, which allowed the incorporation of a transparent thermal link without the use of optical adhesives. These approaches are not without new challenges. The relatively poor thermal conductivity of the YLF thermal link creates a larger than desired temperature gradient between the payload and cooling crystal, which limits the base temperature of the payload. Thermal link materials that have significantly greater thermal conductivity and a CTE that is comparable to the cooling crystal are required to realize the full potential of optical
refrigeration. The silica aerogel supports performed well in the present static application. More studies are needed to assess their thermo-mechanical performance under conditions of mechanical shock and vibration.

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Author contributions
M.P.H., M.S.-R, and R.L.E., conceived the approach and supervised the work. M.P.H. designed the thermal link, and E.R.L. designed the clamshell. J.M., A.G., and A.R.A. performed the laser-cooling experiments at UNM. A.G. first implemented the Herriott cell. C.E.H. synthesized the silica aerogels at LANL. S.P.L. assisted with the FTIR measurements. All authors participated in writing the article. We would like to dedicate this paper to the memory of co-author Eric R. Lee who passed away unexpectedly since the submission of the manuscript. He was instrumental to the success of this work.

Conflict of interest
The authors declare that they have no conflict of interest.

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