**ANALYSIS OF GLOBAL PRACTICES OF FORMATION WATER INTAKE PROFILE STABILIZATIONS BASED ON CROSS-LINKED POLYMER GELS**

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**АНАЛИЗ МИРОВОГО ОПЫТА ПРИМЕНЕНИЯ ТЕХНОЛОГИЙ ВЫРАВНИВАНИЯ ПРОФИЛЕЙ ПРИЕМНОСТИ НА ОСНОВЕ СШИТЫХ ПОЛИМЕРНЫХ ГЕЛЕЙ**

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**ПОЛОЖЕНИЕ НА ОСНОВЕ СШИТЫХ ПОЛИМЕРНЫХ ГЕЛЕЙ**

В в связи с расширением фонда скважин, находящихся на поздней стадии эксплуатации, увеличивается потребность во вводе в эксплуатацию новых комплексов технологий снижения обводненности продукции скважин. В конечном итоге снижение обводненности дает возможность длительное время эксплуатировать обводненный фонтон, повысив выработку остаточных запасов нефти. Наиболее доступным и применимым на производстве решением для решения задач выравнивания профиля приемности пластов является поликарбамид. На основе обобщения международного и отечественного опыта проведен анализ эффективности применения различных модификаций методов с использованием поликарбамидов для снижения обводненности продукции скважин. Рассмотрены такие технологии как: традиционное полимерное заводнение; гели типа in situ; коллоидные дисперсные гели; предварительно сшитые гели. Сшитые полимерные гели представляют собой трехмерные структуры, в которых полимерные цепи связаны либо ионными, либо ковалентными полярными связями. В зависимости от размера частиц они подразделяются на макрогели (от 100 мм до нескольких см) и микрогели (от 0,1 до 30 мм). Применение макро- и микрогелей, в сравнении с составами in situ, в значительной степени снижает риск деформации структурных целей от механических, химических и термических факторов. В результате анализа сформированы рекомендации по использованию различных модификаций закачки полимеров с учетом геолого-технологических условий разработки эксплуатационных объектов. Для снижения обводненности продукции скважин на месторождениях в условиях коллекторов с проницаемостью более 500 мД рекомендуется применение макрогелей; от 100 до 500 мД – микрогелей; от 10 до 100 мД – коллоидных дисперсных гелей. Для повышения эффективности разработки необходима своевременная адаптация перспективных технологий к геолого-технологическим условиям, проведение опытно-промышленных работ на отечественных месторождениях.

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**Ключевые слова:** полимерное заводнение, поликарбамид, профиль пласта, проницаемость, коллоидные дисперсные гели, микрогели, макрогели, частицы полимерного геля, повышение нефтеотдачи.

**Keywords:** polymer water flooding, polyacrylamide, formation profile, permeability, colloidal dispersion gels, microgels, macrogels, polymer gel particles, enhanced oil recovery.
Introduction

One of the most common ways to reduce wellbore fluid watercut in oil wells is to influence the formation water intake profile through injection wells by blocking (reducing) water filtration via the most permeable interbeds. This leads to a change in the filtration flows in the oil formation, involves new oil saturated intervals in the filtration and, eventually, enables a prolonged operation of encroached wells.

Presently, the injection of polymers on the basis of polyacrylamide is the most applicable technology for solving the filtration flow redistribution problems in Russia. This technology has many modifications [1]. Alternative solutions aimed at stabilizing the water intake profiles include the use of cross-linked polymer gels, i.e. three-dimensional structures, in which polymer chains are cross-linked by either ionic or covalent polar bonds. The analysis of literature distinguishes two main groups of the cross-linked polymer gels, such as in situ gel systems and preformed gels [2–4].

The main difference between the in situ and preformed gel technologies is the location of shaping the three-dimensional structure, i.e before or after the injection into the formation. The in situ gels undergo a stage of 3D structure shaping in the formation, whereas the precrosslinked gels form a 3D structure either during synthesis or during a compound makeup immediately prior to the injection into the formation. Fig. 1 shows a classification of polyacrylamide-based compounds used to control the formation water intake profile.

The study analyzes the current experience of using the stabilization technology of the formation water intake profiles based on various modifications of the cross-linked polymer gels, and gives recommendations for application of the most effective methods in different geological and technological conditions of oilfield operations.

Analysis of Technological Problems of Water Soluble Polyacrylamide Applications in Standard Polymer Water Flooding

Polymer water flooding has been used in oil production for more than 50 years. Water soluble polyacrylamide increases water viscosity when dissolved in it. The technology is used in oilfields with high viscosity oil (up to 150 mPa·s), in the temperature range up to 140 ºС, with the formation water salinity up to 270 g/L. In case the formation oil viscosity is increasing, this technology requires an increased viscosity of the injected polymer solution [5]. The injected viscous polymer solution first fills highly permeable intervals of the formation. This causes an increase in filtration resistance, which, in turn, leads to a further redistribution of the new injected liquid flows to the low-permeability interbeds, and displacement of the residual oil. This process is characterized by mobility ratio, which is a ratio of the water phase mobility to the oil phase mobility [6]:

\[
M_r = \frac{k_w \cdot \mu_w}{k_o \cdot \mu_o},
\]

where \(k_w\) and \(k_o\) are the water and oil specific permeability coefficients, \(\mu_o\) and \(\mu_w\) are the oil and water viscosities.

An effective oil displacement process requires the mobility ratio to be below one. With the polymer water flooding, this is achieved by increasing the viscosity of the injected water phase. The polymer solution viscosity is selected on the basis of cost effectiveness and possibility to inject the polymer. The effectiveness of the polymer water flooding depends on the dynamics of rheological parameters of the polymer solution in the formation conditions [7, 8]. The polymer sedimentation in presence of polynvalent metals and polymer adsorption on rock surfaces can significantly reduce the water phase viscosity and, consequently, the water flooding effectiveness. The polymer adsorption process in oil-containing pores can lead to their blocking [9].

During the injection, the polymer solution viscosity can fall due to the three main reasons, i.e. chemical, mechanical and thermal exposures. Chemical destruction of polymers occurs due to redox reactions between the formation water components and possible radical reactions; mechanical degradation occurs in the pumping equipment and at high injection rates; the viscosity of regular polyacrylamide can also fall due to thermal exposure. In order to minimize the polymer
destruction processes, polyacrylamide modifications have been developed, which are described in [10].

Partially hydrolyzed polyacrylamide is the most widely spread polymer modification (Fig. 2, a). The polymer increases the solution viscosity at lower concentrations of the polymer compared to regular polyacrylamide, and is more resistant to chemical destruction. The hydrolyzation degree can vary from 25 to 35 %, whereas the presence of carboxyl groups in the polymer structure make the chain more rigid owing to their mutual repulsion. Introduction of carboxyl groups into the polymer chain does not fully solve the problem of the rheological properties stability. The presence of metal salts in polyacrylamide solution can make its viscosity significantly lower [11]. In order to ensure thermal stability of polyacrylamide in the conditions of high temperature formations, polyacrylamide chain is modified by adding extra monomers increasing its thermal stability, e.g. 2-acrylamido-2-methylpropane sulfonic acid (Fig. 2, b), vinylpyrrolidone etc. Hydrophobically modified polyacrylamide is polyacrylamide chain with integrated hydrophobic monomers (less than 5 % from acrylamide quantity), which can be located at polymer ends or in the chain. In water medium, hydrophobic parts of the polymer gather in groups and form microdomains that stabilize rheological properties of the polymer solution.

Lessons learned from the implementation of the polymer water flooding method suggest that it has certain drawbacks. Polyacrylamide and partially hydrolyzed polyacrylamide are very sensitive to concentrations of metal ions in reservoir water and high formation temperatures (more than 90 °C). Dependence of rheological properties of the solutions on water salinity leads to a mismatch between the actual and planned viscosity value during the technology implementation. Besides, at high injection rates of pumping units, the polymer chains tend to break, which leads to the solution’s viscosity decrease. All chemical modifications of polyacrylamide directed at stabilization of rheological properties can only partially rectify the technology drawbacks, but significantly increase its cost. Apart from the cost of the reagents, it is notable that the technology implementation requires expensive stationary equipment for the makeup of the reagent solutions in a nitrogen atmosphere.

The cross-linked polymer gels help to significantly increase the technology effectiveness by stabilizing the rheological properties and reducing the polymer molecules’ adsorption on the rock surface. Their analysis is given below.

### Analysis of Geological and Technological Conditions for an Effective Use of In Situ Gel Compounds

In situ polymer gel compounds consist of the two main components, i.e. high-molecular-weight polymer and cross-linking agent, the latter is capable of forming bonds with respective polymer chain fragments. The process of shaping into a 3D structure occurs in the formation after injection. The cross-linking agent is added to the polymer solution in the process of makeup at the wellhead. The solution of the polymer and cross-linking agent is called gelant. Such cross-linking agents can be polyvalent metals or organic compounds. The most common inorganic cross-linking agents are aluminum (Al³⁺) and chromium (Cr³⁺) salts; iron (Fe³⁺) and zirconium (Zr⁴⁺) ions are also used [12–14]. Salts of the polyvalent metals dissociate in water into ions and enter into the ionic interaction with carboxyl groups of suspensions of precrosslinked gel particles (PGP).

Inorganic cross-links are recommended at the formation temperatures up to 60 °C. This is associated with a high cross-linking speed, which significantly increases the compound viscosity, making the compound difficult to inject into the formation [15]. At the formation temperature above 80 °C, organic cross-links are used to form heat-resistant bonds between the polymer chains. Covalent bonds are formed between amide groups of the polymer chains and the cross-linking agent. Compared to ionic bonds formed by metal cross-links, the covalent bonds are more resistant to high temperatures [16]. Low compound viscosity during injection and the possibility to control a 3D structure shaping time are among other advantages of the organic cross-links [17]. The following compounds can be used as cross-linking agents:

- phenol and formaldehyde derivatives (specifically, 2,4,6-hydroxymethylphenol) [18]. The cross-link is

![Fig. 2. Structure of modified polyacrylamide: a is partially hydrolyzed; b is sulfonated](image-url)
used in oilfields with the formation temperatures of 70–80 °C in alkaline conditions [19]

- resorcin and urotropin for low-temperature formations with a high formation water salinity [20]
- a blend of pyrocatechin, resorcin and pyrogallol, which has been successfully tested at 150 °C [21]
- polyethyleneimine for cross-linking PAA at 130 °C [22] etc.

In terms of rheological characteristics, in situ polymer gels exist in several varieties depending on the polymer concentration. In [23] they are distinguished into the following basic types: bulk gels, weak gels, colloidal dispersion gels (CDG). Bulk gels are formed at 4,000 mg/L concentration of the partially hydrolyzed polymer. At such a high polymer concentration, the cross-linking agents added to the solution can only form intermolecular bonds, connecting polymer chains into a 3D structure. The cross-linked polymer viscosity can exceed 30,000 mPa·s, while the thermal stability of the bulk gel is limited to 120 °C. Weak gels are formed at polymer concentrations of 800 to 2000 mg/L. Their viscosity can vary in the range from 100 to 10,000 mPa·s, depending on the cross-link type and concentration. Ionic cross-links stabilize the 3D structure up to 80 °C, covalent polar cross-links stabilize the 3D structure up to 100 °C.

The overview [24] about the successful practices of in situ gel use contains examples of treatments that helped to substantially reduce the wellbore fluid watercut. Generally, in situ gels have high penetration ability. Choosing the right cross-linking agent recipe can increase the 3D structure shaping time to 8 h, making the technology useful for blocking encroached channels in the formation uninvaded zone (FUZ). Nevertheless, the following drawbacks of using in situ gel systems can be mentioned:

- the gel disintegrates at high injection rates and in pumping units
- varying molecular weights of polymer chains of the compound cause non-uniformity of the polymer injection into the formation. Thus, polymer chains with a lower molecular weight can be injected faster and farther (chromatographic effect), they can penetrate low-permeability oil interbeds and block filtration in them
- the gel is sensitive to pH, temperature, and the formation water salinity, making it difficult to control mechanical characteristics of the gel in the formation.

The position of the colloidal dispersion gels in the classification of the cross-linked polymer systems varies in different sources. Some authors classify them as in situ gels [2, 4], others [3] classify them as microgels. This type is a transitory polymer structure between in situ gels and preformed gels. Polymer concentration in CDG is only 100–1200 ppm. At such a low content, the polymer chains in presence of a cross-linking agent undergo intramolecular cross-linking, forming globules of a certain shape and size, stable in the range of temperatures from 40 to 94 °C. The polymer concentration in the solution, intended for the colloidal dispersion gel formation, must be below the critical concentration, which is calculated by the formula (2) [25].

\[ C = \frac{M}{\left( N_A (2R_g)^3 \right) }, \]  

where \( M \) is the polymer molecular weight; \( N_A \) is Avogadro number; \( R_g \) is the radius of gyration.

The polymer chains in presence of the cross-links (polyvalent metals) form 1–150 nm globules [26]. Formation of microspheres can be determined by the solution viscosity decrease [27]. Colloidal disperse gels have proven to be highly effective in oilfields of China, USA, Argentina, and Columbia [28–30]. Study [31] proposes the basic parameters of the colloidal gel injection technology implementation in oilfields. It is recommended to plan CDG operations so as to avoid a jump in the injection pressure and polymer drive-out from the well together with the product. Therefore, the injection pressure should be controlled through particle concentration, polymer to cross-link concentration ratio, and injection rate. The reduction of the filtration flow velocity helps to minimize the reagent’s effect on the bottom-hole zone (BHZ) and improve accuracy of the compound injection into the necessary highly permeable interval. Prevention of the reagent drive-out from the formation through the producing well can be achieved through the practice of the preliminary injection of in situ gel into the formation before CDG injection [32]. Formations treatment can be staged, depending on the injection pressure dynamics. At high initial permeability of the formation, more concentrated CDG suspension is used; the increasing injection pressure requires reduction of the gel concentration and particle size without changing the injection pressure [33]. The basic geological and technological conditions of the CDG technology in wells are shown in Table 1.

It is possible to use the colloidal dispersion gels for low permeability formations from 10 mD. Compared to the polymer water flooding or treatment with in situ gel, the CDG technology has a number of advantages, such as a selective treatment of a formation (globules cannot penetrate low permeability oil intervals); better effectiveness at low polymer consumptions; higher polymer capability to enter FUZ. Drawbacks of the colloidal dispersion gel technology are the same as in the case of in situ gels: particle structure destruction by mechanical impacts.
The depletion period of the producing wells.

The section, thus they increase the productivity and redistributed to less permeable uninvaded intervals of interbeds. As a result, filtration flows are in water and filling the most permeable encroached absorbents, i.e. they are capable of limited swelling in water and filling the most permeable encroached interbeds. As a result, filtration flows are redistributed to less permeable uninvaded intervals of the section, thus they increase the productivity and depletion period of the producing wells.

The precrosslinked polymer gels are polymer particles of a certain size distributed in a dispersion medium, thus forming a suspension. These polymers were designed to rectify drawbacks of in situ technology; specifically, to reduce the polymer destruction under the effect of mechanical and chemical factors. Inherently, polymer particles are absorbents, i.e. they are capable of limited swelling in water and filling the most permeable encroached interbeds. As a result, filtration flows are redistributed to less permeable uninvaded intervals of the section, thus they increase the productivity and depletion period of the producing wells.

The precrosslinked gels can be divided into two main types: macrogels (preformed particle gels (PPG)) and microgels (Bright Water (thermotropic gels)). A choice between the reagent types is based on the technical and geological conditions of specific production sites.

Macrogels are particles obtained using the method of bulk polymerization with further drying and mechanical fragmentation down to the desired fraction size. Particle sizes for compounds classified as macrogels range between 100 µm to several cm. Particles of this size in the English-language literature are commonly termed as the preformed particle gel (PPG). PPG particles are characterized with the absorption capacity (a weight of water absorbed by a unit weight of a polymer) in the formation water from 1 to 100 g/g, the stability in saline water, and the thermal stability at 140 °C during more than one year. Morphological and absorption characteristics can be controlled by changing a polymer recipe [36, 37].

By now, the PPG technology has been used in deposits with high watercut formations for more than 20 years. In the native oilfields, this technology is limited to developing polymer compounds and pilot production operations. So, study [38] contains results of successful filtration tests of the PPG reagent adapted to the conditions of oilfields in the Volga-Ural oil and gas province (formation temperature below 30° and high salinity of the formation water).

Globally, the PPG technology presently accounts for more than 4000 well operations [39]. Many successful instances are known when the PPG technology was used in high temperature and low temperature formation conditions, with various degrees of salt content in the formation water, in carbonate reservoirs and terrigenous formations. The basic limitation is applicability of the PPG technology only in conditions of reservoirs with permeability above 500 mD. Injection of macrogels into less permeable reservoirs can lead to BHZ blocking and, consequently, well shutting-in. This fact significantly reduces the potential extent of production sites where PPG could be used in the native oilfields. For instance, in Perm Krai only about 60 production sites, which is about 10 % of all wells, have reservoir permeability exceeding 500 mD. Thus, for most oil wells, the PPG technology needs adaptation. For medium permeability reservoirs, the cross-linked polymer gels with higher filtration characteristics are required.

### Analysis of Geological and Technological Conditions for Effective Use of Cross-Linked Polymer Compounds Based on Macrogels

Precrosslinked polymer gels are polymer particles of a certain size distributed in a dispersion medium, thus forming a suspension. These polymers were designed to rectify drawbacks of in situ technology; specifically, to reduce the polymer destruction under the effect of mechanical and chemical factors. Inherently, polymer particles are absorbents, i.e. they are capable of limited swelling in water and filling the most permeable encroached interbeds. As a result, filtration flows are redistributed to less permeable uninvaded intervals of the section, thus they increase the productivity and depletion period of the producing wells.

The analyzed international experience of using the cross-linked polymer gels to reduce formation watercuts shows that in the conditions of reservoirs with 100 to 500 mD permeability, microgels with a particle size from 0.1 to 30 µm have the highest effectiveness (Fig. 3). Fig. 3, a, shows a suspension of microgels with a concentration of 100 µg/L, where the size of swollen particles amounts to 25 µm. Fig. 3, b, shows the distribution of swollen microgel particles (size 10–100 µm) in pore channels with throats of 200–1000 µm. Fig. 3, b, shows how particles group in the pore space, completely filling it [43].

The cross-linked microgels can be obtained using two basic methods, the method of explosive polymerization and inverse emulsion polymerization. The explosive polymerization method is simple and cost-effective, but difficult in terms of controlling sizes and morphological characteristics of particles.

### Table 1

| Parameter                     | Indicator       |
|-------------------------------|-----------------|
| Formation characteristics     |                 |
| Temperature, °C              | 25–100          |
| Permeability, mD             | 10–4200         |
| Oil viscosity, cPs           | 5–30            |
| Initial pressure, atm        | 0–95            |
| Treatment parameters         |                 |
| Concentration during treatment, ppm | 250–1200        |
| Cross-link                   | Chromium acetate (8–31) |
| Polymer to cross-link ratio  | 20:1 to 80:1    |
| Injection speed, m³/day      | 24–320          |
| Maximum injection pressure, atm | 50–150         |
| Injection volume, m³/well    | 1600 … > 104 000 |

ISSN 2224-9923. Perm Journal of Petroleum and Mining Engineering. 2020. Vol.20, no.2. P.150-161
The inverse emulsion method resolves the aforementioned drawbacks and simplifies the control of a particle size during synthesis [40]. Polymer particles obtained through the inverse emulsion method have a size of 0.1 to 50 µm and are capable of swelling five-fold. The maximum absorption capacity is achieved in 10–50 days depending on the particle structure and formation temperature [41–45]. Swelling kinetics and absorption capacity of microspheres also depend on certain formation water salinity. Microgels of this type are widely used in Chinese oilfields (Daqin, Jidong, Dagan, Shengli) [46, 47].

There are several varieties of microgels, e.g. standard cross-linked, temperature sensitive (Bright-water), pH-sensitive. The experience of using standard microgels in Jidong field (the formation temperature is 90 °C, the formation water salinity is 5 g/L) is described in [47]. In the conditions of the pronounced formation inhomogeneity (from 10 to 1000 mD), the water intake profile stabilization was performed using microgels with a mean particle size in the dry state 21 µm (scatter 0.4 to 50 µm). The reagent injection was performed in two stages. Initially, to block the largest channels, microspheres with sizes of 20 to 50 µm were injected in the concentration of 2000 mg/L; at a later stage, particles with sizes of 0.4–30 µm and suspension concentration of 1500 µm were injected. The microgel mother solution was supplied through a dosing pump during the injection. During 65 days of wells treatment using this technology, the injection pressure has gradually grown from 10 to 14 MPa, which confirms the injection effectiveness. Eventually, the obtained operational benefit was substantial, as the wellbore fluid watercut decreased from 43 to 28 %, on average [47].

Bright-water is an instance of the temperature sensitive microgels, the product jointly developed by BP, Chevron, Texaco, and Nalco. Polymer particles of the Bright-water microgel have the size of 0.1–1 µm in a dry condition, and high penetration ability. For this reagent, the formation temperature triggers the process of sharp and irreversible swelling of particles, thereby strongly reduces the reservoir permeability. The microgel particles swelling mechanism is controlled by the proportion of the two types of the polymer cross-links: stable and unstable. The quantity of the stable cross-links in a particle is 200–600 times more than the quantity of the unstable cross-links, since they ensure polymer stability and prevent its dissolution in the formation water. The size of the swollen particles and swelling speed (type and quantity of cross-links) has to be comparable with the size of reservoir pores. The recommended geological and technological conditions of the technology applications are as follows: highly permeable porous reservoir, minimum fracturing, temperature from 50 to 150 °C, water salinity less than 7 %.

Study [48] contains a description of a pilot production operations based on the Bright Water technology at Milne Point field (Alaska) on one injection well and two producing wells. Initially, the injection of particles was performed at a high rate in order to push them through BHZ. After exiting BHZ, the particles filled the most permeable intervals of the formation. Under the effect of the formation temperature (50–75 °C) the temporary cross-link was destroyed and swelling of the particles occurred. The injection well absorbed nearly 270 m³ reagent per day at a pressure of 150 atmospheres during 21 days. Overall, 60.8 tons of the polymer were injected at a concentration of 3300 ppm (0.33 %) in the water. The resulting oil gain for the first producing well was obtained nine months after the treatment, for the second well it was eleven months after the treatment. The operational profitability effect from the treatment lasted for about two years [48].

Generally, it can be concluded that the use of microgels versus in situ compounds significantly reduces the risks of the polymer chain destruction by mechanical, chemical and thermal factors, which is caused by the 3D polymer structure. Unlike the PPG technology, the microgels can increase the technology application scope manifold due to their ability to effectively enter the formation at permeability values starting from 100 mD.

Conclusion

The analysis of lessons learned from applying various polymer technologies has shown that a series of promising methods is successfully implemented abroad to reduce watercut of wells at advanced stages of operation. The data were used to develop recommendations concerning the selection of the precrosslinked polymer gel technology taking into account the reservoir permeability (Table. 2).
The innovative technology, which is recommended in order to reduce the wellbore fluid watercut in reservoirs with permeability more than 500 mD consists in using macrogels. As for reservoirs with permeability from 100 to 500 mD, it is advisable to use microgels, while the colloidal dispersion gels are recommended for reservoirs with permeability from 10 to 100 mD. To improve the development effectiveness, the advanced technologies need to be adapted to certain geological and technological conditions of pilot production operations in native oilfields.

Acknowledgements

The work has been supported by the Russian Ministry of Education and Science within the Federal Targeted Programme for Research and Development in Priority Areas of Development of the Russian Scientific and Technological Complex for 2014-2020 (Unique project ID RFMEFI62120X0038), and by Perm Krai Administration within funding under the Grant for Scientific Research of International Research Groups (Project C-26/174.7).

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Please cite this article in English as:
Ketova Iu.A., Galkin S.V., Votinov A.S., Kang W., Yang H. Analysis of the international practice in application of conformance control technologies based on cross-linked polymer gels. Perm Journal of Petroleum and Mining Engineering, 2020, vol.20, no.2, pp.150-161. DOI: 10.15593/2224-9923/2020.2.5

Просьба ссылаться на эту статью в русскоязычных источниках следующим образом:
Анализ мирового опыта применения технологий выравнивания профилей приемистости на основе сшитых полимерных гелей / Ю.А. Кетова, С.В. Галкин, А.С. Вотинов, В. Канг, Х. Янг // Вестник Пермского национального исследовательского политехнического университета. Геология. Нефтегазовое и горное дело. – 2020. – Т.20, №2. – С.150–161. DOI: 10.15593/2224-9923/2020.2.5