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Functionalized Mesoporous Silica Nanoparticles as Delivery Systems for Doxorubicin: Drug Loading and Release

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Abstract: Functionalized nanoparticles have played a major role in the field of targeted therapy, owing to their ability to control the release and for the selective delivery of entrapped materials to tumours. In this work, we described the loading capacity and in vitro release kinetics of mesoporous silica nanoparticles (MSNs), functionalized with Poly-L-Histidine and Tamoxifen. The model drug Doxorubicin (DOX) was successfully encapsulated into MSN-based systems, using the technique of solvent immersion. A post-surface grafting loading method was investigated on functionalized systems, with DOX loading content determined using HPLC. Dialysis bag diffusion was employed to investigate the release kinetics of DOX-loaded-systems at pH 7.4 and 5. The amount of DOX released from native MSNs systems over a 72 h period at pH 5 was approximately 40%; and at pH 7.4 ≈ 30%. A moderate pH dependent release behaviour was observed with both our functionalized systems: DOX@MSN-PLH and DOX@MSN-PLH-TAM; with approximately 5% of DOX released from DOX@MSN-PLH-TAM at pH 7.4 and about 9% released at pH 7.4 over 72 h. The maximal cumulated release of DOX molecules from DOX@MSN-PLH after 72 h was ≈18% at pH 7.4 and ≈23% at pH 5, respectively. The outcome of this work offers a promising contribution towards building future stimuli-responsive nano-drug delivery systems.

Keywords: mesoporous silica nanoparticles; doxorubicin; tamoxifen; Poly-L-Histidine; drug delivery

1. Introduction

The recent trends of using nanotechnology to resolve unmet medical needs have been recognized as a Key Enabling Technology in the European Union (EU) [1]. By developing advanced drug delivery systems (DDSs), therapeutic compounds of interest can be entrapped within nanoparticle matrices or conjugated to different nanosystems [2,3], drug molecules can be delivered to designated molecular targets at much lower doses, and over longer time periods [2–4]. The development of nanoformulations has revolutionized the delivery and targeting of therapeutic compounds, providing an economic and innovative utilization of existing drugs [5–9].

Numerous nanocomposites have received considerable attention as promising DDSs. Mesoporous silica nanoparticles (MSNs), discovered in 1992 by the Mobile Oil Corporation, have emerged as one of the most advanced and versatile nano DDSs [10–13]. Small sized MCM-41 MSNs are the most frequently used carriers for pharmaceutical applications, owing to their unique mesoporous structure, high surface area, large pore volume and tuneable pore diameter [10–15]. In addition to this, MSNs also provide a solid framework with abundant surface chemical moieties, allowing surface functionalization with various ligand molecules, for the targeted and controlled release of encapsulated drug moieties to a particular site [10,11,14–16].
Even though MSNs has been widely used in drug delivery, considerable efforts towards novel compositions of MSN-based systems have been attempted through surface engineering with different targeting functionalities, to enhance the accuracy and selectivity of these systems as nanocarriers, beyond that of unfunctionalized MSNs. As a result, recently we presented an exciting conceptual study in which a novel functionalized MSN-based system was synthesized and characterized (Figure 1a) [16]. The fabricated MSNs are highly uniformed in spherical shapes and sizes (≈150 nm), with well-ordered porous channels, without the interconnection between each individual channel (Figure 1b). This represented the first step of our journey into the development of a nano modality with the ability to transport anticancer agents selectively to breast tumours, to reduce off-target adverse effects. In that work, the pH-sensitive polymer Poly-L-Histidine (PLH) and the FDA-approved hormonal drug Tamoxifen (TAM) were appended to native MSNs. Our functional nanostructure is believed to lay a foundation for a new theoretical concept of triple-action DDSs, which includes: (i) passive targeting using nanosized MCM-41 as a DDS; (ii) active targeting through the selective binding of TAM to estrogen receptors (ERs)-overexpressing-breast tumours; (iii) controlled drug release with PLH molecules acting as a pH sensitive gatekeeper, preventing premature drug leakage [16].

![Figure 1.](image)

**Figure 1.** (a) The triple-action nanostructure DDS based on MSNs, functionalized with the pH sensitive PLH polymer and the hormonal drug Tamoxifen. Image adapted from Day et al. [16]. (b) TEM images demonstrating the diameter and internal structure of MSNs on the 200 nm scale-bar, demonstrating well-defined ordered long-range structure of inner mesopores and highly ordered porous channels.

A high drug loading capacity (the mass ratio of drugs to nanosystems) is an important parameter to determine whether a nanostructure can be employed as an ideal DDS [17]. In this study, we will load our previously synthesized functionalized nanostructures with Doxorubicin (DOX), to investigate loading ability of our system. DOX has been proven to be effective against both early and metastatic breast cancer, owing to its ability to interfere with DNA and RNA synthesis in solid tumours [18–20]. However, the potential clinical applications of DOX are hindered by its highly toxic profile and non-specific mechanisms of action which can cause fatal damage to healthy and sensitive tissues, leading to cardiac arrest in some cases [19,20]. Therefore, successful loading of DOX onto MSN-based systems
is expected to enhance its delivery, via both passive diffusion and active targeting with functionalized MSNs.

There are two general methods by which drug molecules such as DOX are loaded onto functionalized MSNs systems: (i) pre-surface grafting [21–27] (Figure 2a) and (ii) post-surface grafting (Figure 2b) which is less frequently applied [28,29]. For the pre-surface grafting method as illustrated in Figure 2a, native MSNs are immersed into a stirred solution of DOX, the DOX molecules then diffuse into the porous channels of the MSNs. The resulting DOX@MSNs (in which @ means loaded) are then grafted with PLH and TAM, forming DOX@MSN-PLH-TAM (Figure 2a). For the post-surface grafting method, fully functionalized MSN-PLH-TAM will be immersed into a stirred solution of DOX (Figure 2b). The resulting DOX@MSN-PLH-TAM particles are then collected by centrifugation. In this study, we are particularly interested in the loading of DOX into our novel MSN-based systems (Figure 1), using the unconventional post-surface grafting loading approach (Figure 2b). This is expected to avoid the leakage of DOX molecules which may occur during each step of surface functionalization with the pre-surface grafting method.

![Figure 2. Illustration of the pre-grafting loading of MSN-based system.](image)

Additionally, in this study, we focused on the in vitro release kinetics of our functionalized systems, attributed to the grafted PLH molecules. With the theory that PLH molecules which have a pK_a ~6.10, under physiological conditions (pH 7.4), will condense on the surface of MSNs. This will block the MSN pores and prevent the release of DOX [16]. Subsequently, as the PLH-gated nanosystems enter into tumours, the PLH molecules will become protonated by the acidic tumour environment (pH ~5). The cationic PLH molecules, due to charge repulsion, are then expected to extend out from the surface of the MSNs, thus opening the pores to release DOX [16]. In addition to this, it was also hypothesized that the greater solubility of DOX molecules in acidic environments (pH 5) [30,31] in combination with the introduction of PLH molecules onto MSNs should enhance the release of entrapped DOX molecules in acidic tumour microenvironments.

2. Materials and Methods

2.1. General Methods and Materials

TAM was obtained from Combi-Blocks Inc. (San Diego, CA, USA). PLH (Mal-PLH-NH_2/Mal-GH_26KK, purity 90%, 26 units, Mw = 4.05 kDa) was purchased from Leon
Biological Technology Co. Ltd. (Nanjing, China). Doxorubicin.HCl was obtained from Sapphires Bioscience Ltd. (Adelaide, Australia). All other reagents were purchased from Sigma Aldrich and used without further purification. All pH measurements were conducted on a pre-calibrated Orion Ross pH meter (Metrohm AG, Herisau, Switzerland). RP-HPLC Chromatography was performed on a Shimadzu HPLC system (Kyoto, Japan) equipped with a PDA detector (start $\lambda$ 200 nm, end $\lambda$ 350 nm, analysed at 250 nm), using a C18 column (4.6 × 150 mm, 5 µm) as the stationary phase. The mesostructures and diameters of MSN-PLH-TAM were analysed by high-resolution transmission electron microscopy (HR-TEM) (Hitachi S-4800, Tokyo, Japan) obtained on an HC electron microscope at an accelerating voltage of 100 kV.

2.2. Synthesis of Native MSN, MSN-PLH and MSN-PLH-TAM

The synthesis of three MSN-based systems; namely native MSNs, MSN-PLH and MSN-PLH-TAM directly followed our recently published study [16]. Native MSNs were synthesized via the Stöber method, using CTAB as the templating structure and TEOS as the silica precursor. The surface (OH-) moieties of MSNs were converted to thiols using MPTES and conjugated to Mal-PLH-NH$_2$ molecules via Michael addition at pH 5, to give MSN-PLH. Resulting MSN-PLH nanostructure was then reacted with mesylated TAM-TEG building blocks prepared in-house (in which TEG- Tetraethylene glycol was used as an organic linker) to give the requisite MSN-PLH-TAM nanomaterial.

2.3. Loading of DOX onto MSN-Based Systems

The solvent immersion technique for loading of native MSNs was adapted and modified from a previous study by Hakeem and colleagues [32]. MSNs (5 mg) were ultrasonically dispersed in 5 mL solution of DOX in MilliQ water (DOX conc = 1 mg mL$^{-1}$) for 15 min. The suspension remained under stirring for 24 h at room temperature, protected from light. DOX@MSN was collected using centrifugation (12,000 RPM, 30 min, 24 ºC). The supernatant was collected separately; while solid DOX@MSNs were washed with MilliQ water (2 mL × 3) until supernatant became clear, indicating the absence of free DOX particles in the wash. HPLC was then employed to quantify the amount of free/unloaded DOX molecules in the combined supernatants. The HPLC analytical method was adapted and modified from a previously published procedure [33]. From the free DOX determined by HPLC, loaded DOX can be quantified based on the initial amount of DOX in the original solution. A calibration curve was obtained from a serial dilution of stock solution containing pure DOX in MilliQ water (concentration 1 mg mL$^{-1}$). Encapsulation efficiency (EE%) and drug content (DC%) were determined using the following Equations (1) and (2), respectively [34,35].

\[
EE\% = \frac{\text{Weight of loaded DOX}}{\text{Initial weight of DOX}} \times 100\%
\]

\[
DC\% = \frac{\text{Amount of loaded DOX}}{(\text{Weight of MSNs} + \text{Weight of loaded DOX})} \times 100\%
\]

A previous procedure from Bilalis et al. [36] was adapted and modified for the post-surface grafted loading of MSN-PLH and MSN-PLH-TAM (Figure 2b). In short, MSN-PLH or MSN-PLH-TAM (15 mg) were ultrasonically dispersed in 6.25 mL of PBS buffer (pH 5) containing DOX (concentration 1.5 mg mL$^{-1}$). The resulting solutions were then stirred vigorously (400 RPM) at room temperature and protected from light for 24 h. DOX@MSN-PLH and DOX@MSN-PLH-TAM were collected by centrifugation (12,000 RPM, 20 min, 24 ºC), and washed with water (2 mL × 3). Total supernatant and washing solvents were recovered and HPLC was then employed to quantify EE% and DC% of both systems following Equations (1) and (2) above.

2.4. In-Vitro Release Kinetics

To investigate the in vitro release pattern of DOX@MSN, DOX@MSN-PLH and DOX@MSN-PLH-TAM, a dialysis diffusion technique was employed. To mimic systemic
circulation where pH is 7.4, and the tumour environment where pH is ≈ 5; PBS buffer solutions at each of the respective pHs were employed as releasing media. Then, 1 mg of each sample was dispersed with buffer solution (1 mL) and placed in cellulose membrane dialysis bag, which was then placed in 10 mL of respective release media at 37 ± 0.5 °C, with 100 rpm paddle rotation. At 2, 4, 6, 24, 28 and 72 h, aliquots of sample (1 mL) were withdrawn and replenished with the same volume of fresh media. Pure DOX in buffers was used as positive controls, while the suspensions of native MSNs in buffers were used as negative control.

3. Results and Discussion

3.1. Drug Loading

For the pre-surface grafting method (Figure 2a), when DOX is loaded into the native MSNs, the colour of particles changed from white to red, indicating the presence of DOX molecules in the sample. According to the data obtained by HPLC, DOX@MSNs achieved an EE of 7.78 ± 0.18%, and a DC of 7.22 ± 0.153%. This was in accordance with the original data by which this procedure was adapted, reporting an EE of ~6.5% by weight [32]. Even though in their original study, Hakeem et al. did not report their DC result; the obtained DC value of ~7% was also in accordance with other literature procedures which have reported DC values of less than 10% [36]. No extra signals were observed in the FT-IR of the DOX@MSN following encapsulation (see Figure S1 in Supplementary Material). This suggests that the loaded DOX content might be just too low to be detected by FT-IR.

Following the loading of DOX into the MSNs, the DOX@MSNs was then functionalized with PLH and TAM as per our previous work (see Supplementary Material for full characterization) [16]. The synthesis is a continuous/one-pot reaction, in which the resulting DOX@MSN-PLH was immediately reacted with TAM-TEG-OMs, to yield DOX@MSN-PLH-TAM. It was assumed that the maintenance of red colour indicated that the DOX molecules remained inside the porous channels during the grafting steps. This is as per previous literatures [37,38], and no obvious red colour of the washing solutions was observed. Therefore, at the end of surface grafting, DOX@MSN-PLH-TAM was assumed to have an EE of 7.78 ± 0.18, similar to that of DOX@MSNs. This assumption was considered during the in vitro release experiments.

The colour change upon DOX loading was also observed for our functionalized MSNs, MSN-PLH and MSN-PLH-TAM; where the colour of the particles changed from white to red upon loading (Figure 3a shows the colour change for the MSN-PLH-TAM system). In Figure 3b, the FT-IR spectra of DOX@MSN-PLH-TAM demonstrated the same peaks as the unloaded MSN-PLH-TAM which was in accordance with our previously published study [16]. No additional signal was obtained; this suggests that the loaded DOX content might be just too low to be detected by FT-IR. This was also consistent for the MSN-PLH vs. DOX@MSN-PLH system (see Figure S2 in Supplementary Material).

DOX@MSN-PLH achieved an EE of 12.2% while DOX@MSN-PLH-TAM achieved a similar EE of 12.1%. The EE of these functionalized systems was slightly greater than that of DOX@MSN. This may be due to enhanced interactions between the DOX molecules and the hydrophobic PLH molecules due to π-π stacking [36], leading to an increased amount of DOX becoming encapsulated. Interestingly, this increased trend was not observed with the DC of both functionalized systems. DOX@MSN-PLH and MSN-PLH-TAM achieved a DC of ≈7%, which is very similar to that of the native MSNs.
3.2. Drug Release

The next step in the development of our nanosystem was to characterize the release of DOX. This was done using a diffusion bag technique, where each MSN-based system was loaded into a dialysis bag and the release media collected at varying time points. For this experiment a positive control was employed, where pure DOX was loaded into the dialysis bag and the releasing media consisted of either the pH 7.4 or pH 5 buffer systems. The media collected was then analysed by HPLC, in which the characteristic peak of pure DOX was obtained at 4.3 min, while the peak eluted before three minutes were the solvent (buffers) peak as per previous literature [33].

The first systems to be evaluated were those synthesized by the pre-surface grafting loading method (Figure 2a). When the DOX@MSNs from the pre-grafting loading method were evaluated for release, DOX was detected in both the release mediums of pH 7.4 and 5 (Figure 4). This indicated that encapsulated DOX was liberated into the releasing media. In addition to this, the peak of DOX from DOX@MSN at pH 7.4 has a lower AUC compared to the peak of DOX from DOX@MSN at pH 5, indicating a higher release of DOX from native MSNs under acidic conditions. This is as expected as DOX is known to be more soluble at acidic pH [31,32].

Unfortunately, when the release of the functionalized particles DOX@MSN-PLH-TAM (synthesized by pre-surface grafting method, Figure 2a) was evaluated using the same release procedure, no DOX was detected in the release media. It was hypothesized that there was insufficient amount of DOX left within the mesopores after multiple surface grafting steps and that what remained was well below the detection limit of the instrument. This is consistent with the similar FT-IR spectrum of the DOX loaded systems vs. unloaded systems, implying a low loading of DOX (Figure 3; Figures S1 and S2).

To investigate the in vitro release pattern of DOX@MSN-PLH and DOX@MSN-PLH-TAM (synthesized by post-surface grafting loading method, Figure 2b), the release of DOX from native MSNs was used as a control for comparison. The release kinetic of DOX@MSNs, DOX@MSN-PLH and DOX@MSN-PLH-TAM were demonstrated in Figure 5 below.
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More DOX was released at acidic pH in all three systems; native MSNs, MSN-PLH and MSN-PLH-TAM. This reaffirmed the knowledge that DOX molecules are more soluble in acidic environments [31,32]. For native MSNs, 8.7% more DOX was released at pH 5 than at pH 7.4. It was observed that the release of DOX@MSNs reached a plateau, with an overall release after 72 h of 33.6% for DOX at pH 7.4; and 42.3% at pH 5. However, the total amount of DOX released from the native MSNs was less than typically expected from this type of material, where a release of over 80% after 24 h is reported [30,37]. This incomplete release from native MSN may be attributed to absorption of DOX molecules inside the deep channels of our materials, requiring the need for releasing media to penetrate inside and solubilize the drug, hence, prolonging the releasing time.

For functionalized MSNs, it was observed that the cumulative amount of DOX released from MSN-PLH and MSN-PLH-TAM was less than that of native MSNs. The maximal cumulated release of DOX molecules from DOX@MSN-PLH after 72 h was ≈18% at pH 7.4 and ≈23% at pH 5, respectively. Whereas approximately 5% of DOX was released from DOX@MSN-PLH-TAM at pH 7.4 and 9% released at pH 5 over 72 h (Figure 5). Due to the complexity of the synthesis for functionalized systems, the release experiments of DOX@MSN-PLH and DOX@MSN-PLH-TAM (synthesized by post-surface grafting approach, Figure 2b) were only conducted once for each sample. However, the pilot study still confirmed a decrease in DOX release from DOX@MSN-PLH and DOX@MSN-PLH-TAM system, compared to that of native DOX@MSNs. This implied a potential pore blocking effect due to the addition of surface functionalities. Additionally, this may be due to the (-OH) groups of encapsulated DOX molecules forming some additional hydrogen bonds with amine groups (from the lysine moiety) of the PLH molecules and/or with the oxygen atoms of TAM molecules.

We observed a pH sensitive release behaviour of DOX in our novel nanostructure MSN-PLH-TAM, with more DOX released at acidic pH compared to that of biological pH. However, the release of encapsulated DOX from MSN-PLH-TAM decreased at both pHs, compared to that of MSN-PLH. This may suggest that obtained delayed release was a result of the pore-blocking effect caused by the increased surface moieties, following the conjugation of TAM-TEG building blocks onto MSN-PLH. In addition to this, it is possible that some hydrophobic stacking between TAM and PLH molecules may have
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4. Conclusions

DOX was successfully encapsulated into three MSN-based systems, namely MSN, MSN-PLH and MSN-PLH-TAM, using a post grafting loading method. The release kinetics in all three systems demonstrated a pH-sensitive pattern, with more DOX released at acidic pH. Following the attachment of TAM molecules into the MSN-PLH, the pH-effect was still retained. Despite the lower amount of DOX released from our functionalized system as compared with native MSNs, our developed functionalized MSNs were able to demonstrate an effect of pH on the release kinetics. Even though obtained data demonstrated expected pH-sensitive patterns of the functionalized systems, more in vitro pharmacodynamics investigations need to be performed, to confirm if this effect is sufficient to offset the dose and drug-associated toxicities of DOX on healthy cells, while still maintaining its efficacy against tumours. Our future work will improve the release of DOX, as well as investigate the targeting capacity of our systems against breast cancer with cellular uptake studies.

Supplementary Materials: The following are available online at https://www.mdpi.com/article/10.3390/app11136121/s1, Figure S1: FT-IR spectrum of DOX@MSNs vs unloaded native MSNs, Figure S2: FT-IR spectrum of DOX@MSN-PLH vs unloaded MSN-PLH.

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