Disposable Pipette Extraction Phase Based on Styrene–Divinylbenzene/Pernigraniline Composite, Applied for Dexamethasone Determination in Synovial Fluid by HPLC with UV Detector

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A reliable method using disposable pipette extraction (DPX) based on composite of pernigraniline and styrene–divinylbenzene (Sty–DVB) copolymer was applied to the analysis of dexamethasone in synovial fluid using high-performance liquid chromatography and ultraviolet (UV) detector (HPLC–UV).

DPX variables, namely, number of draw/eject cycles for extraction or desorption, sample pH, volume, and desorption solvent, were optimized to establish the best sorption equilibrium and analysis time. The highest extraction efficiency value was obtained with 50 μL of synovial fluid mixed with 1950 μL of water, in five cycles of 300 μL of sampling, followed by liquid desorption of the drug with 300 μL of methanol in three cycles. The developed method demonstrated a linear response over the range from 10 to 100 ng/mL, with $R^2 = 0.993$. The limit of quantification (LOQ) was 10 ng/mL. Based on the validation results, the proposed method can be a useful tool to detect dexamethasone levels in synovial fluid.

Keywords: Dexamethasone, DPX/HPLC–UV, synovial fluid, pernigraniline

Introduction

Dexamethasone is a corticosteroid drug that is usually involved with the proteins synthesis rate controlling.

Its main effect is the profound change made in lymphocyte immune response due to anti-inflammatory and immunosuppressive action also preventing or suppressing inflammatory processes of various natures. It is found in the pharmaceutical market as tablets, aerosol, creams, and ophthalmic suspensions [1]. Its empirical formula is $C_{22}H_{29}FO_{5}$ and its structural formula is represented in Figure 1. It is rapidly absorbed after oral administration, and up to 65% of a dose is excreted in the urine within 24 h. Its pharmaceutical effects strongly depend on its distribution in tissues and body fluids, such as synovial fluid. Synovial fluid is present in the synovial joints; it is a thick and stringy fluid, which reduces friction between articular cartilage and other tissues and joints, cushioning and lubricating them during the movements. For a better understanding of these effects, a detailed investigation of its pharmacokinetics is extremely necessary.

The most commonly used techniques for dexamethasone analysis have been liquid chromatography and spectrophotometric methods, as described in international official compendiums, depending on the pharmaceutical forms [2].

Liquid–liquid extraction (LLE) [2–5] and solid-phase extraction (SPE) [6, 7] are the most frequently sample preparation techniques used for drug extraction from biological fluids. These techniques are laborious and time consuming and require large amounts of organic solvents. The trends in analytical chemistry point toward methods that lead to simplification, miniaturization of analytical instrumentation and sample preparation techniques, and reduction of organic solvent and sample volumes [8].

A dispersive solid-phase micro extraction technique, known as disposable pipette extraction (DPX), relies on a disposable pipette tip with sorbent loosely contained inside of it, separated by a lower screen and upper barrier, enabling mixing the sorbent with sample solutions. One of its great advantages is that these tips do not require conditioning once the solution is mixed with the sorbent. In addition, channeling and flow rates have no impact on extraction efficiency for this type of dispersive solid-phase extraction technology.

DPX was developed as an alternative to traditional SPE, combining efficient and rapid extraction with significantly reduced solvent and time consumption. The extractions can be fully automated, thus reducing the risk of human error, improving repeatability, and increasing throughput [9, 10].

Figure 1. Structural representation of dexamethasone ($pK_a = 12.4$)

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The first commercially available micropipette tip was based on chromatographic media, C18 microparticulates; since then, different phases with different interaction modes have been introduced [11, 12].

Recently our group presented a DPX extraction phase based on styrene-divinylbenzene (Sty–DVB) copolymer and polyaniline (PANI) composites, which was suitable for the antidepres-
sants (fluoxetine and norflutoxetine) in plasma samples by high-
performance liquid chromatography and fluorescence detection (HPLC–FD) [8]. The developed DPX method is shown to be very rapid, taking just a few minutes to perform without any solvent evaporation. However, higher standard deviation was observed between the developed phases suggesting poor chemical stability of the composites. To overcome this problem, Sty–
DVB/PANI synthesis protocol has been improved.

In this paper, we describe a simple, sensitive, and reliable DPX/HPLC–UV method for the rapid determination of dexamethasone in a synthetic fluid using a Sty–DVB/pernigraniline composite as the extraction phase. Polyaniline (PANI) can exist in three different redox forms, such as leucoemeraldine base (the most reduced form of PANI), emeraldine salt (the 50% intrinsi-
cally oxidized form, in the salt form), and pernigraniline (the most oxidized form of PANI) [13].

**Experimental**

**Chemicals and Samples.** 1.4-Dioxane (UV/HPLC grade), benzoyl peroxide (BP) 65%, acetone 99.5%, aniline PA, and hydrochloric acid 37% were obtained from Vetec (Rio de Janeiro, Brazil). Toluene 99.5% and heptane 98% were obtained from LABIMPEX (São Paulo, Brazil). Methanol 99.8%, ethanol 99.5%, and gelatin powder were obtained from SYNTH (São Paulo, Brazil). The hydroxyethylcellulose (HEC) was acquired from Polithechno (São Paulo, Brazil), NaCl 99% from SYNTH (São Paulo, Brazil). Toluene 99.5% and heptane 98% were obtained from ISOFAR (Rio de Janeiro, Brazil). Styrene (Sty) was obtained from Lyondellbasell (Dallas, TX, EUA), and benzoyl peroxide (BP) 65%, acetone 99.5%, aniline PA, and hydrochloric acid 37% were purchased from PARCHEN (New York, U.S.A.), both were obtained from ISOFAR (Rio de Janeiro, Brazil), styrene (Sty) was obtained from Lyondellbasell (Dallas, TX, EUA), and divinylbenzene (DVB), as a mixture of 50:50 m- and p-isomers, was purchased from PARCHEN (New York, U.S.A.), both were purified by reduced pressure distillation.

The dexamethasone analytical standards were donated by Lilly (São Paulo, Brazil). The working standard drug solutions were prepared by diluting the stock solutions of these drugs (1 mg/mL in methanol) to a proper volume of methanol, based on their therapeutic intervals. These solutions were stable for 45 days at −20 °C. Water purified in a Milli-Q system (Millipore, São Paulo, Brazil) was used to prepare the mobile phase. Synovial fluids without dexamethasone, used as blank samples, were donated by the FarmaTec (Centro de Pesquisa, Desenvolvi-
mento e Inovação Tecnológica em Fármacos, Medicamentos e Cosméticos) of Universidade Federal de Goiás. These studies were performed in accordance with the World Medical Asso-
ciation’s “an Ethical principle for medical research involving human and animal’s subjects”. These synovial samples were spiked with target drug and used to optimize the DPX process and validate the analytical method.

**Synthesis of Styrene–Divinylbenzene (Sty–DVB) Copolymer.**

Sty–DVB copolymer synthesis was carried out following the methodology described in a previous work [14]. Briefly, the aqueous phase (AP) was composed by HEC at 0.45% (v/v), NaCl 0.60% (v/v), and gelatin at 0.12% (v/v). The organic phase (OP) was prepared dissolving 1% of initiator BP, Sty at 26% (v/v) and DVB at 84% (v/v) monomers. The porogenic agents were heptane and toluene in a volume ratio of 85:15 and 150% dilution degree in relation to monomers volume. The ratio AP/OP was kept at 4.1 (v/v). The OP was added to the AP, the temperature was kept at 70 °C with stirring at 600 rpm for 24 h. Finally, the copolymer beads were filtrated and washed with water and then with ethanol to remove synthesis residues. The copolymer used to prepare the composite was sieved in the range of 53–75 μm.

**Synthesis of Sty–DVB Copolymer/Pernigraniline Composite.**

To Sty–DVB copolymer/pernigraniline composite synthesis, 1 gram of Sty–DVB copolymer was put in contact with 10 mL of acetone/aniline solution with a volume ratio of 20:80, and stirring for 3 h. A reactive solution was prepared by mixing 0.14 g of BP in 5 mL of dioxane, 1.5 mL of water, and 1.25 mL of hydrochloric acid P.A. The swollen copolymer was filtrated and added to the reactive solution. The pernigraniline polymerization was carried out under stirring, at 25 °C for 24 h. Afterwards, the composite was filtrated and washed with methanol and dried under ambient conditions. The complete procedure was repeated 4 times to ensure the maximum rate of reaction.

**Physical–Chemical Characterization FTIR.** Fourier transform infrared (FTIR) spectra were recorded using a Perkin Elmer Spectrum Frontier in the range of 4000–4000 cm⁻¹ for Sty–DVB copolymer and Sty–DVB/pernigraniline composite. Samples were analyzed using the attenuated total reflectance, ATR.

**Scanning Electron Microscopy (SEM).** Morphology of Sty–
DVB/Pernigraniline composite was evaluated by SEM, in a Jeol JSM – 6610 Thermoscientific NSS Spectral Imaging (Peabody, MA, USA).

**Nitrogen Adsorption Measurements.** The specific surface area and pore size distribution measurements were performed using a Micrometrics ASAP 2010 nitrogen sorption porosimeter. Analysis was via nitrogen sorption carried out at 77 K. Specific surface area was determined by BET method and the pore size distribution by the BJH method based on nitrogen desorption isotherm.

**Chromatographic Conditions.** The HPLC–UV analyses were performed on a Young-Ling, YL-9100 (Korea) chromatographic system equipped with a UV-diode array detector (λ = 205 nm) and a HT800L autosampler (Gloucester, UK). The separation was performed in a Lichrosphere 60° RP: Select B (250 mm × 4 mm, 5 μm particle size) (Merek, Darmstadt, Germany) column at temperature of 25 °C, the chromatographic runs were performed in isocratic mode with a water–acetonitrile (50:50, v/v) mobile phase, at flow rate of 1.0 mL/min.

**Optimization of DPX Process.** DPX extraction was carried out in a pipette tip of 1 mL, containing 50 mg of Sty–DVB–
Pernigraniline composite. The sample consisted in a mixture of water/synovial fluid placed in a sample vial, in a ratio of 1950 μL:50 μL water–synovial fluid. The synovial fluid was spiked with dexamethasone standard solution at therapeutic levels (25 ng/mL). To establish the sorption equilibrium and shorten the analysis time, several parameters (number and volume of draw/eject sampling cycles, sample pH, type, and volume of desorption solvent, and number of draw/eject desorption cycles) were optimized.

The influence of the sample pH on the extraction efficiency was evaluated at different pH values ranging from 4.0 to 8.0 (0.05 mol/L buffer solutions). Different solvents (water, acetonitrile, methanol, water–methanol [50:50], and acetonitrile–water [50:50]). The used desorption solvent volume (100–700 μL) was also evaluated to establish the desorption conditions. After the desorption process, the sorbent was washed with 500 μL of water–methanol solution (50:50, v/v). The carryover was also evaluated.

**Analytical Validation.** In order to evaluate the linearity and all responses for the selected parameters, an analytical curve was constructed plotting the dexamethasone peak areas vs. concentration (ng/mL of dexamethasone in synthetic fluid) and

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applying linear regression to it. Accuracy and inter-assay precision were determined by means of quintuplicate assays of the synovial fluid samples spiked with dexamethasone solution representing the entire range of the calibration curve. Accuracy values were calculated by the differences between concentrations of dexamethasone added to the synovial fluid samples and those obtained after extractions, by means of the analytical curve.

Results

DPX Extraction Phase Development. As mentioned on introduction section, the synthesis of Sty–DVB/PANI composite was based on our previously published work [8], however the developed composite in that case, did not exhibit adequate chemical stability and reproducibility between extractions cycles. In order to improve the extraction phase, the synthesis protocol has been modified resulting in the pernigraniline composite which has an over oxidation state. The morphology of synthesized Sty–DVB/pernigraniline composite was characterized by SEM. As shown in Figure 2, the composite exhibited a spherical and porous structure, of small particles which suggests high surface area and large extraction capacity. Such characteristic was confirmed by nitrogen adsorption analysis for Sty-DVB/pernigraniline composite, which showed a surface area of 323 m²/g, a total pore volume of 1.34 cm³/g, and an average pore diameter of 10 nm.

Figure 3 shows the FTIR spectra for copolymer Sty–DVB and composite Sty–DVB/pernigraniline. The pernigraniline characteristic band located at 1488 cm⁻¹ could be attributed to the stretching of the quinone (Q). The absence of 1556 cm⁻¹ band, from benzenic (B) rings, is a property of pernigraniline [15], and the shoulder at 1630 cm⁻¹ and the bands at 1445, 1414, and 1374 cm⁻¹ are present in the spectra of the products obtained at low acidity. Other characteristic of pernigraniline could be attributed to the absence of band at 1310 cm⁻¹, resulting from symmetrical stretching of CN groups of secondary aromatic amines. The band at 1159 cm⁻¹ is attributed to the CH group, which is due to nitrogen linked to the quinone group [15, 16].

The extraction ability of pernigraniline could be attributed to the interactions between polymer and analyte, such as base–acid, dipole–dipole interactions, and hydrogen bonding, in addition to the π–π and hydrophobic interactions. The increased affinity of pernigraniline and dexamethasone can be attributed to π–π and hydrophobic interactions.

The physical and chemical stability of the composite has been investigated. The composite was subjected to a wide pH range (1 to 11), salt concentrations (NaCl in 1%, 5%, and 10% m/V concentrations), and different solvents composition (methanol, water, acetonitrile, and mixtures of methanol–water [50:50, v/v]). The composite was stable for all investigated conditions. Comparing with commercial DPX phases (silica based), the Sty–DVB/pernigraniline was chemically stable at higher pH values such as 11.0, while the silica derivate phases are damaged by strongly alkaline solutions [10–12, 17].

DPX Conditions Optimization. To reaching new extraction phases and methods, the sorption capacity, method sensitivity, and carryover are important parameters to be considered. Using the composite Sty–DVB/pernigraniline, and the DPX conditions were evaluated based on the extraction efficiency represented by chromatogram dexamethasone peak area.

The number of draw/eject cycles is a critical parameter for extraction recovery. The dexamethasone recovery rate increased above 1 extraction cycles (1 × 300 μL, Figure 4). The extraction cycles were performed with the same aliquot, using five extraction cycles. This condition results in suitable limit of quantification (LOQ) value (10 ng/mL), precision (7.6 to 10.2%), and accuracy (81 to 93%) for dexamethasone at therapeutically levels. Therefore, this condition (5 × 300 μL), was performed in less than 30 s, so it was used for subsequent experiments.

The sample pH was evaluated, and the obtained results suggested that in basic pH solutions, the electrostatic repulsion between analyte and pernigraniline coating was observed, which resulted in lower extraction efficiencies. Among the evaluated pH sample values, higher sensitivity of the DPX/HPLC–UV method was obtained after the sample was diluted with 1950 μL of water, in which the drug sorption into Sty–DVB/pernigraniline (Figure 5) had been favored.

Figure 2. SEM micrographs of the Sty–DVB/Pernigraniline composite at 40-fold (A) and 500-fold (B) magnification

Figure 3. IR spectra of copolymer Sty–DVB (A) and composite Sty–DVB/Pernigraniline (B)
Analyte was eluted by drawing/ejecting the solvent through the pipette tip. Methanol, water, acetonitrile, acetonitrile–water (50:50), and methanol–water (50:50) were evaluated as desorption solvents. Among these, methanol exhibited the highest desorption efficiency using three desorption cycles of 300 μL for the same aliquot (3 x 300 μL) (Figure 6).

The extraction efficiency increased, comparing one to three draw/eject cycles (300 μL), but above three cycles, the number of cycles did not substantially change the results (Figure 7). To improved recovery method values, methanol was drawn and remained in contact with the extraction phase for 10 s prior to being dispensed.

Between the extractions, the extraction phase was washed with methanol for 3 draw/eject cycles (500 μL) to ensure that the matrix compounds were removed. A pipette tip containing the extraction phase was used over 20 times with water solutions without extraction efficiency decrease, and for the synovial fluid, each pipette tips was used 10 times with no extraction efficiency decrease was observed.

Based on these data, the best DPX experimental conditions among those investigated for dexamethasone assays (Figures 4–7) were 50 μL of synovial fluid mixed with 1950 μL of water, five extraction cycles (5 x 300 μL), followed by liquid desorption of the analyte with 300 μL methanol in three desorption cycles.

**Analytical Validation.** As shown in Figure 8A and B, the developed method was selective, once in the drug-free synovial fluid sample (Figure 8B), there is no significant signal comparing to the spiked sample (Figure 8A) with dexamethasone at the therapeutic concentration (25 ng/mL).

According to Figure 8, all those analytical signals relative to the possible interference at the same analyte retention time were lower than 20% of the chromatographic signal of the target drug at the concentration corresponding to the quantification limit concentration.

The linearity of the DPX/HPLC–UV method ranged from the LOQ of 10 ng/mL to 100 ng/mL for dexamethasone. The linear regression provided the equation $y = 9.47x + 251.94$ with $R^2 = 0.993$ ($x$ being the dexamethasone concentration, in ng/mL, and $y$ represents the peak area obtained from chromatogram, in mV s). All points for analytical curve were performed in replicate ($n = 5$). The limit of quantification (LOQ) was attributed as the lowest concentration on the analytical curve for which the coefficient of variation (CV) was close to 10% (Table 1), based on a signal-to-noise ratio of 10. The calculated LOQ was 10 ng/mL.

The accuracy and inter-assay precision of the DPX/HPLC–UV method (5 replicates) were evaluated by using synovial fluid samples spiked with dexamethasone at different concentrations (Table 1). The method precision was calculated according to the
CV% (inter-assay) at three levels. The CV% values ranged from 7.6% to 10.2% (Table 1), and the relative recovery was assessed via analysis of synovial fluid samples spiked with standards at the concentrations showed in Table 1. The recoveries were calculated by comparing the UV peak areas of the spiked samples with the peak area of the direct injection of dexamethasone solution at the same concentration, achieving values ranging from 54 to 81%.

The complete DPX process described here, from the conditioning, sample loading, and elution, till washing steps, needed only 3 min. This represents a huge reduction compared to traditional SPE methods, which can last for approximately 20 min [12]. The low synthesis cost, high recoveries, and the ability to reuse the same phase until 20 times without loss in efficiency are additional advantages to this new Sty–DVB/pernigraniline DPX extraction phase.

Table 1. Inter-day precision (CV), accuracy, and recovery of DPX/HPLC–UV method for dexamethasone in synovial fluid analysis

| Concentration (ng/mL) | CV (%) | Accuracy (%) | Recovery (%) |
|-----------------------|--------|--------------|--------------|
| 10                    | 10.2   | 81           | 54           |
| 25                    | 9.8    | 85           | 69           |
| 75                    | 7.6    | 93           | 81           |

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Conclusion

A new DPX extraction phase based on Sty–DVB/pernigraniline composite was successfully developed and applied to the analysis of dexamethasone in synovial fluid, showing good linearity and reproducibility under optimized conditions.

According to the miniaturization trends in analytical chemistry, the small volumes of synovial fluid and solvents, combined with reduced preparation time and cost, ensure that the presented work represents important advances not only to the sample preparation subject, but also for the environmental matter.

The method presented here represents a powerful tool for rapid, accurate, and quantitative determination of dexamethasone in clinical analyses.

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