A rigorous derivation of multicomponent diffusion laws

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Abstract

This article is concerned with the dynamics of a mixture of gases. Under the assumption that all the gases are isothermal and inviscid, we show that the governing equations have an elegant conservation-dissipation structure. With the help of this structure, a multicomponent diffusion law is derived mathematically rigorously. This clarifies a long-standing non-uniqueness issue in the field for the first time. The multicomponent diffusion law derived here takes the spatial gradient of an entropic variable as the thermodynamic forces and satisfies a nonlinear version of the Onsager reciprocal relations.

Keywords. Multicomponent diffusion laws, conservation-dissipation structure, Maxwell iteration, Onsager reciprocal relations

1 Introduction

Multicomponent diffusion plays a dominant role in many industrial and natural processes and has been extensively studied in a large number of chemical engineering books and articles since 1948. See [1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14, 15] and the references cited therein. A fundamental task of these studies is to determine the relations between the thermodynamic fluxes and forces, which is called multicomponent diffusion laws or constitutive relations. In the literature, various different constitutive relations can be found and they are not always consistent with each other. For example, in [9] published in 2006 one can find the statement “The fact that $D_{ij}$ (diffusion coefficients) is fundamentally nonunique is clearly documented in the literature.1−4,14−18”. Therefore, there has been a urgent need for a mathematical clarification of the situation.

In this paper, we present a mathematically rigorous derivation of a multicomponent diffusion law for a mixture of gases and thereby clarify the uniqueness for the first time. For the sake of simplicity, we assume that all the gases are isothermal and inviscid. Our derivation follows the Furry-Williams approach [1, 3, 9, 10] starting from the macroscopic balance equations of all the gases and uses the Chapman-Enskog expansion or Maxwell iteration recently justified in [16] for a class of hyperbolic relaxation systems. The justification is crucially based on a conservation-dissipation structural property of the system of the macroscopic equations. The conservation-dissipation structure was firstly proposed and was shown in [17].
to be satisfied by many classical models from mathematical physics. Recently, it has been successfully used in [18] to develop a general theory of mathematical modeling of irreversible processes.

The multicomponent diffusion law derived here takes the spatial gradient of an entropic variable as the thermodynamic forces and satisfies a nonlinear version of the Onsager reciprocal relations. Namely, the corresponding diffusion coefficient matrix is symmetric and positive-definite. The use of the entropic variable as thermodynamic forces is a key of the original Onsager relation [19] and provides us a nice formulation of the multicomponent diffusion laws. This formulation is mathematically and numerically important in treating the multicomponent diffusion systems.

As mentioned earlier, there are various multicomponent diffusion laws in the literature and it is almost impossible to comment on all of them. For most of the existing multicomponent diffusion laws, the diffusion coefficient matrix is generally non-symmetric [10, 11] and sometimes even not completely determined [9]. A further comparison of ours with that in [9] is given in Section 4. On the other hand, the Maxwell iteration was also used in [12, 13] to derive the multicomponent diffusion laws. We also notice [14, 15] where the multicomponent diffusion laws were derived in two different frameworks of non-equilibrium thermodynamics. Unfortunately, all these derivations are formal, lack a mathematically rigorous justification, and therefore the uniqueness issue has been not resolved before.

The paper is organized as follows. In Section 2, we introduce the classical mass and momentum equations for multicomponent diffusion systems. Section 3 focuses on the conservation-dissipation structure of the systems. The multicomponent diffusion law is derived in Section 4.

## 2 Governing equations

Consider a mixture of \( N \) gases. Assume that all the gases are inviscid and isothermal. Then the motion of the gases obeys the classical mass and momentum equations [9, 10]. Let \( \rho_i \) and \( \mathbf{V}_i \) denote the density and mass averaged velocity of species \( i \), respectively. The species-specific mass equations read as

\[
\frac{\partial \rho_i}{\partial t} + \nabla \cdot (\rho_i \mathbf{V}_i) = \dot{W}_i, \quad i = 1, \cdots, N, \tag{2.1}
\]

where \( \dot{W}_i \) represents the net mass production rate (per unit volume) of species \( i \) due to chemical reactions. Since mass can either be created or destroyed by chemical reactions, the \( \dot{W}_i \)'s always satisfy

\[
\sum_{i=1}^{N} \dot{W}_i = 0.
\]

On the other hand, the momentum equation for species \( i \) is

\[
\frac{\partial (\rho_i \mathbf{V}_i)}{\partial t} + \nabla \cdot (\rho_i \mathbf{V}_i \mathbf{V}_i + p_i I_d) = \rho_i \mathbf{f}_i + \dot{W}_i \mathbf{V}_i + \mathbf{G}_i^{\text{coll}}, \quad i = 1, \cdots, N. \tag{2.2}
\]

Here \( p_i \equiv p_i(\rho_i) \) is the pressure of species \( i \), \( I_d \) is the unit matrix of order \( d \) \((d = 1, 2, 3)\), \( \rho_i \mathbf{f}_i \) is the body force acting on species \( i \), and \( \mathbf{G}_i^{\text{coll}} \) stands for the net impacts of all interspecies collisional momentum exchanges on species \( i \). Since interspecies collisions conserve the total momentum, the sum of \( \mathbf{G}_i^{\text{coll}} \) over
all species must be identically zero:

\[ \sum_{i=1}^{N} G_i^{\text{coll}} = 0. \]  

(2.3)

By its definition, \( G_i^{\text{coll}} \) is a function of \( X = (V_1 - V_i, \ldots, V_N - V_i) \) and vanishes whenever \( X = 0 \).

The latter means that there is no collisional momentum exchange with species \( i \) when other species all have the same velocity. Based on this observation, we may write

\[ G_i^{\text{coll}} = \int_0^1 \frac{\partial G_i^{\text{coll}}}{\partial X}(\theta X) d\theta \cdot X \equiv \sum_{j \neq i} K_{ij}(V_i - V_j) \]

with \( K_{ij} = [K_{ij}^{\alpha\beta}]_{d \times d} \) being \( d \times d \)-matrixes. \( K_{ij}^{\alpha\beta} \) is called a collisional coefficient, possibly depending on any quantities like densities under consideration.

For the Stefan-Maxwell model \[1, 3, 9\], each \( K_{ij} \) is a scalar matrix, that is,

\[ K_{ij}^{\alpha\beta} = -\sigma_{ij} \delta_{\alpha\beta}, \quad \sigma_{ij} = \bar{m}_{ij} \nu_{ij}. \]

Here \( \delta_{\alpha\beta} \) is the Kronecker delta, \( \bar{m}_{ij} = m_i m_j / (m_i + m_j) \) is the “reduced mass”, \( m_i \) is the molecular mass of species \( i \), and \( \nu_{ij} \) is the averaged frequency (per unit volume) of collisions between molecules \( i \) and \( j \), and satisfies

\[ \sigma_{ij} = \sigma_{ji} > 0 \]

for each \( i \neq j \) and \( i, j = 1, \ldots, N \).

For future references, we rewrite \( G_i^{\text{coll}} \) as

\[ G_i^{\text{coll}} = -\sum_{j=1}^{N} K_{ij} V_j \]

with

\[ K_{ij} = \delta_{ij} \sum_{k=1}^{N} \sigma_{ik} - \sigma_{ij} \]

for each \( i, j = 1, \ldots, N \). Obviously, \( K_{ij} \equiv K_{ij}(\rho_1, \rho_1 V_1, \ldots, \rho_N, \rho_N V_N) \) satisfies

\[ \sum_{i=1}^{N} K_{ii} \equiv 0 \quad \text{and} \quad K_{ij} = K_{ji} < 0, \quad i \neq j, \]  

(2.4)

for \( i, j = 1, \ldots, N \). This is consistent with (2.3).

In a mixture of gases, collisions usually happen much faster than the macroscopic fluid motions. Denote by \( \varepsilon \) the ratio of characteristic collision time to characteristic fluid mechanics time. Then \( \varepsilon \) is small and the collision coefficient \( K_{ij} \) may be scaled as

\[ K_{ij} = \frac{1}{\varepsilon} \tilde{K}_{ij} \]  

(2.5)

with \( \tilde{K}_{ij} = O(1) \). For the notational convenience, we will use \( K_{ij} \) for \( \tilde{K}_{ij} \) in the rest of this paper.
3 Conservation-dissipation structure

In this section, we show that the system of equations (2.1) and (2.2) with fast collisions (2.5) possesses the conservation-dissipation structure proposed in [17]. To do this, we rewrite (2.1) and (2.2) as

\[
\frac{\partial U_i}{\partial t} + \sum_{j=1}^{d} \frac{\partial \hat{F}_j(U_i)}{\partial x_j} = \frac{1}{\varepsilon} Q_i(U), \quad i = 1, 2, \ldots, N
\]

or

\[
\frac{\partial U}{\partial t} + \sum_{j=1}^{d} \frac{\partial F_j(U)}{\partial x_j} = \frac{1}{\varepsilon} Q(U)
\]

(3.1)

with the external forces \(\rho i f_i\) and chemical reaction terms \(\dot{W}\) being ignored to simplify the exposition. In (3.1), \(d\) is the space dimension,

\[
U_i = \left( \begin{array}{c} \rho_i \\ \rho_i V_i \end{array} \right), \quad \hat{F}_j(U_i) = \left( \begin{array}{c} \rho_i V_{i,j} \\ \rho_i V_i V_{i,j} + p_i e_j \end{array} \right), \quad Q_i(U) = \left( \begin{array}{c} 0 \\ -\sum_{k=1}^{N} K_{ik} V_k \end{array} \right),
\]

\(U = (U_1, U_2, \ldots, U_N)^T\), \(F_j(U) = (\hat{F}_j(U_1), \ldots, \hat{F}_j(U_N))^T\), \(Q(U) = (Q_1(U), \ldots, Q_N(U))^T\), \(V_{i,j}\) denotes the \(j\)-th component of velocity \(V_i\), and \(e_j\) stands for the \(j\)-th column of the unit matrix \(I_d\).

Following [20], we define a mathematical entropy \(\eta i\) for species \(i\):

\[
\eta_i(\rho_i, \rho_i V_i) = \rho_i \int_{\rho_i}^{\bar{\rho}_i} \frac{p_i(z)}{z^2} dz + \frac{1}{2\rho_i} |\rho_i V_i|^2
\]

with \(\bar{\rho}_i\) a possible positive value. Consequently, a mathematical entropy \(\eta(U)\) for the mixture can be introduced as

\[
\eta(U) = \sum_{i=1}^{N} \rho_i \int_{\rho_i}^{\bar{\rho}_i} \frac{p_i(z)}{z^2} dz + \sum_{i=1}^{N} \frac{1}{2\rho_i} |\rho_i V_i|^2.
\]

(3.2)

The conservation-dissipation structure for system (3.1) can be stated as follows.

**Theorem 3.1** Assume that (2.4) holds true and \(p_i = p_i(\rho_i)\) is strictly increasing for \(\rho_i > 0\) and for each \(i = 1, \ldots, N\). Then the mathematical entropy \(\eta(U)\) defined in (3.2) for system (3.1) is strictly convex in \(O_U = \{U \in \mathbb{R}^{dN} | \rho_i > 0, i = 1, \ldots, N\}\). Moreover, the following three statements are true:

- \(\frac{\partial^2 \eta(U) \partial F_j(U)}{\partial U^2}\) is symmetric for each \(U \in O_U\) and for each \(j\);
- There exists a symmetric and semi-positive matrix \(L(U)\) such that
  \(Q(U) = -L(U) \frac{\partial \eta(U)}{\partial U}\);
- The null-space of \(L(U)\) is independent of \(U \in O_U\).

**Remark 3.1** The first statement is the well-known entropy condition for hyperbolic conservation laws and corresponds to the classical principles of thermodynamics. The second one can be understand as a
nonlinearization of the celebrated Onsager reciprocal relation in modern thermodynamics [21] and implies the second law of thermodynamics. It displays a direct relation of irreversible processes to the entropy change. The last one expresses the fact that physical laws of conservation hold true, no matter what state the underlying thermodynamical system is in (equilibrium, non-equilibrium, and so on).

Here is our proof of the above theorem. Compute

\[
\frac{\partial^2 \eta(U)}{\partial U_i \partial U_k} = \left( \frac{\partial \eta(U)}{\partial \rho_i} \right) \left( \frac{\partial \eta(U)}{\partial \rho_k} \right), \quad \frac{\partial^2 \eta(U)}{\partial U_i \partial U_j} = \frac{1}{\rho_i} \left[ \begin{array}{cc} p_i'(\rho_i) + |V_i|^2 & -V_i^T \\ -V_i & I_d \end{array} \right] \delta_{ij}.
\]

Because \( p_i'(\rho_i) > 0 \) for \( \rho_i > 0 \), one can directly verify that the Hessian \( \frac{\partial^2 \eta(U)}{\partial U^2} \) is positive definite for each \( U \in O_U \). Therefore, \( \eta = \eta(U) \) is strictly convex in \( O_U \).

Next we compute from the definition of \( F_j(U_j) \) that, for \( i \neq k \),

\[
\frac{\partial^2 \eta(U)}{\partial U_i \partial U_k} = 0, \quad \frac{\partial F_j(U_j)}{\partial U_k} = 0,
\]

and

\[
\frac{\partial^2 \eta(U)}{\partial U_i^2} \frac{\partial F_j(U_j)}{\partial U_i} = \frac{1}{\rho_i} \left[ \begin{array}{ccc} V_{i,j}|V_i|^2 - p_i'(\rho_i)V_i^T e_j & p_i'(\rho_i)e_j^T - V_{i,j}^T & V_{i,j} I_d \\ V_{i,j} & I_d & \end{array} \right].
\]

Therefore, \( \frac{\partial^2 \eta(U)}{\partial U^2} \frac{\partial F_j(U)}{\partial U} \) is symmetric.

Now we recall the structure of \( Q(U) \) and define a \((d+1)N \times (d+1)N\) matrix \( L(U) \equiv [L_{ik}(U)]_{N \times N} \) with

\[
L_{ik}(U) = \left( \begin{array}{cc} 0 & 0 \\ 0 & K_{ik} I_d \end{array} \right).
\]

Then \( Q(U) \) can be written as

\[
Q(U) = -L(U) \frac{\partial \eta(U)}{\partial U}.
\]

From (2.4) it is not difficult to see that \( L(U) \) is symmetric and nonnegative-definite matrix.

It remains to show that the null-space of \( L(U) \) is independent of \( U \). To do this, we write \( w \in R^{(d+1)N} \) as \( w = \left( \begin{array}{c} w_1 \\ \vdots \\ w_N \end{array} \right) \) with \( w_k = \left( \begin{array}{c} w_k^I \\ w_k^{II} \end{array} \right) \) and \( w_k^{II} \in R^d \). If \( L(U) \cdot w = 0 \), then

\[
\sum_{k=1}^N L_{ik} w_k = \sum_{k=1}^N \left[ \begin{array}{cc} 0 & 0 \\ 0 & K_{ik} I_d \end{array} \right] \left( \begin{array}{c} w_k^I \\ w_k^{II} \end{array} \right) = \left( \begin{array}{c} 0 \\ \sum_{k=1}^N K_{ik} w_k^{II} \end{array} \right) = 0
\]

for each \( i \). Since the null-space of the \( N \times N \)-matrix \( K = [K_{ij}] \) is spanned by \( \{(1, \cdots, 1)^T\} \), it follows that

\[
w_1^{II} = w_2^{II} = \cdots = w_N^{II}.
\]
Therefore, the null-space of $L(U)$ is
\[{\{w \in R^{(d+1)N} : (w^1, w^2, \cdots, w^N) \in R^N \text{ and } w^I_1 = w^I_2 = \cdots = w^I_N \in R^d}\},\]
which is independent of $U$. This completes the proof.

4 Multicomponent diffusion laws

In this section, we derive a multicomponent diffusion law by using the Maxwell iteration [12, 13] or Chapman-Enskog expansion justified in our recent paper [16]. The conservation-dissipation structure provided a highly efficient framework for the justification.

To begin with, we introduce the mass diffusion flux (also called thermodynamic flux)
\[J_i \equiv \rho_i (V_i - V),\]
where the mass-averaged velocity $V$ is defined as
\[V \equiv \frac{1}{\rho} \sum_{i=1}^N \rho_i V_i, \quad \rho \equiv \sum_{i=1}^N \rho_i.\]
From these definitions, it follows immediately that
\[\sum_{i=1}^N J_i = 0.\]
This is called zero-net-flux condition in some literature [9]. Due to this condition, we only need to consider the first $(N - 1)$ fluxes $J_i (i = 1, \cdots, N - 1)$.

Referring to [16], we rewrite the multicomponent diffusion system (3.1) in term of the new variable
\[W \equiv \begin{pmatrix} \rho, \rho V, \rho_1, \cdots, \rho_{N-1}, J_1, \cdots, J_{N-1} \end{pmatrix}^T \leftarrow U \equiv \begin{pmatrix} \rho_1 V_1, \cdots, \rho_N V_N \end{pmatrix}^T.\]
Firstly, the $N$ mass equations are equivalent to
\[
\begin{align*}
\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho V) &= 0, \\
\frac{\partial \rho_i}{\partial t} + \nabla \cdot (\rho_i V + J_i) &= 0, \quad i = 1, \cdots, N - 1,
\end{align*}
\]
which is derived by summing up the mass equations [2.4] over $i$ and using the definition $J_i = \rho_i (V_i - V)$. Recall that the external forces and reaction sources have been neglected. In order to obtain the equations for the total momentum $\rho V$ and the first $(N - 1)$ fluxes $J_i$, we recast the left-hand side of the momentum
equation (2.2) as
\[
\frac{\partial (\rho_i V_i)}{\partial t} + \nabla \cdot (\rho_i V_i \otimes V_i) + \nabla p_i \\
= \frac{\partial J_i}{\partial t} + \frac{\partial (\rho_i V)}{\partial t} + \nabla \cdot (\rho_i V \otimes V) + \nabla p_i + \nabla \cdot \left( J_i \otimes V + V \otimes J_i + \frac{J_i \otimes J_i}{\rho_i} \right).
\]

Summing up these momentum equations over \(i\), we use (2.3) and the zero-net-flux condition to obtain
\[
\frac{\partial (\rho V)}{\partial t} + \nabla \cdot (\rho V \otimes V) + \nabla \sum_{j=1}^{N} p_j + \sum_{j=1}^{N} \nabla \cdot \left( \frac{J_j \otimes J_j}{\rho_j} \right) = 0. \tag{4.2}
\]

Moreover, we use \(\sum_{j=1}^{N} K_{ij} = 0\) in (2.4) and rewrite the collision term
\[
G_{i}^{\text{coll}} = -\frac{1}{\varepsilon} \sum_{j=1}^{N} K_{ij} (V_j - V) \\
= -\frac{1}{\varepsilon} \left( \sum_{j=1}^{N-1} K_{ij} \frac{J_j}{\rho_j} + K_{iN} \frac{J_N}{\rho_N} \right) \\
= -\frac{1}{\varepsilon} \sum_{j=1}^{N-1} K_{ij} \left( \frac{J_j}{\rho_j} - \frac{J_N}{\rho_N} \right) \\
= -\frac{1}{\varepsilon} \sum_{j=1}^{N-1} K_{ij} \left( \frac{1}{\rho_j} \delta_{jl} + \frac{1}{\rho_N} \sum_{l=1}^{N-1} J_l \right) \\
= -\frac{1}{\varepsilon} \sum_{j,l=1}^{N-1} K_{ij} \left( \frac{1}{\rho_l} \delta_{jl} + \frac{1}{\rho_N} \right) J_l.
\]

Thus, the equation for \(J_i\) reads as
\[
\frac{\partial J_i}{\partial t} + \frac{\partial (\rho_i V)}{\partial t} + \nabla \cdot (\rho_i V \otimes V) + \nabla p_i + \nabla \cdot \left( V \otimes J_i + J_i \otimes V + \frac{J_i \otimes J_i}{\rho_i} \right) \\
= -\frac{1}{\varepsilon} \sum_{j,l=1}^{N-1} K_{ij} \left( \frac{1}{\rho_l} \delta_{jl} + \frac{1}{\rho_N} \right) J_l. \tag{4.3}
\]

Consequently, system (3.1) has been rewritten as (4.1)–(4.3).

On the other hand, we recall (2.4) that \(K_{ij} = K_{ji} < 0\) with \(i \neq j\) and \(\sum_{j=1}^{N} K_{ij} = 0\). Then \([K_{ij}]_{(N-1) \times (N-1)}\) is a strictly diagonally dominant and symmetric matrix. Therefore, it has an inverse, say \(K\), and the inverse is symmetric and positive definite. Moreover, we set \[13\]
\[
\Phi_{ij} = \frac{1}{\rho_j} \delta_{ij} + \frac{1}{\rho_N} \quad \text{and} \quad C_{ij} = \rho_j \delta_{ij} - \frac{\rho_i \rho_j}{\rho}.
\]
for $i, j = 1, \cdots, N - 1$. Note that

$$(C\Phi)_{ij} = \sum_{l=1}^{N-1} C_{il} \Phi_{lj} = \sum_{l=1}^{N-1} (\rho_l \delta_{il} - \frac{\rho_l \rho_i}{\rho}) (\frac{1}{\rho_j} \delta_{ij} + \frac{1}{\rho_N})$$

$$= \delta_{ij} + \frac{\rho_i}{\rho_N} - \frac{\rho_i}{\rho} \frac{\rho_N}{\rho N} \sum_{l=1}^{N-1} \rho_l$$

$$= \delta_{ij} + \frac{\rho_i}{\rho_N} - \frac{\rho_i}{\rho} (\rho - \rho_N)$$

$$= \delta_{ij}.$$

and

$$\frac{\partial (\rho_i \mathbf{V})}{\partial t} + \nabla \cdot (\rho_i \mathbf{V} \otimes \mathbf{V}) = \rho_i (\mathbf{V}_t + \mathbf{V} \cdot \nabla \mathbf{V}) + \rho_t + \nabla \cdot (\rho_i \mathbf{V})$$

$$= \rho_i (\mathbf{V}_t + \mathbf{V} \cdot \nabla \mathbf{V}) - (\nabla \cdot \mathbf{J}_i) \mathbf{V}$$

due to the second line in (4.3). The flux equation (4.3) can be rewritten as

$$\mathbf{J}_i = -\varepsilon \sum_{k,l=1}^{N-1} C_{ik} K_{kl} \left( \rho_i (\mathbf{V}_t + \mathbf{V} \cdot \nabla \mathbf{V}) + \nabla p_l - (\nabla \cdot \mathbf{J}_l) \mathbf{V} + (\mathbf{J}_l)_{ij} + \nabla \cdot (\mathbf{V} \otimes \mathbf{J}_l + \mathbf{J}_l \otimes \mathbf{V} + \frac{J_l}{\rho_l} \otimes \mathbf{J}_l) \right)$$

$$= -\varepsilon \sum_{k,l=1}^{N-1} C_{ik} K_{kl} \left( - \frac{\rho_i}{\rho} \nabla \sum_{j=1}^{N} p_j + \nabla p_l + O(\varepsilon) \right).$$

(4.4)

The second step is due to (4.2) and the Maxwell iteration, using that $\mathbf{J}_i = O(\varepsilon)$ indicated by the first step.

Furthermore, we deduce from (4.4) that

$$\mathbf{J}_i = -\varepsilon \sum_{k,l=1}^{N-1} C_{ik} K_{kl} \left( - \sum_{j=1}^{N-1} \frac{\rho_j \rho_l}{\rho} \nabla p_j - \frac{\rho_N \rho_l}{\rho} \nabla p_N - \frac{\rho_l \nabla p_l}{\rho} \right) + O(\varepsilon^2)$$

$$= -\varepsilon \sum_{k,l=1}^{N-1} C_{ik} K_{kl} \left( \sum_{j=1}^{N-1} (\rho_j \delta_{ij} - \frac{\rho_j \rho_l}{\rho} \nabla p_j) \nabla p_l - \frac{\rho_l}{\rho} (\rho - \sum_{j=1}^{N-1} \rho_j) \frac{\nabla p_N}{\rho N} \right) + O(\varepsilon^2)$$

$$= -\varepsilon \sum_{k,l=1}^{N-1} C_{ik} K_{kl} \left( \sum_{j=1}^{N-1} (\rho_j \delta_{ij} - \frac{\rho_j \rho_l}{\rho} \nabla p_j) \nabla p_l - \sum_{j=1}^{N-1} (\rho_j \delta_{ij} - \frac{\rho_j \rho_l}{\rho} \frac{\nabla p_N}{\rho N}) \right) + O(\varepsilon^2)$$

$$= -\varepsilon \sum_{k,l,j=1}^{N-1} C_{ik} K_{kl} C_{lj} \left( \frac{\nabla p_j}{\rho_j} - \frac{\nabla p_N}{\rho_N} \right) + O(\varepsilon^2)$$

for $i = 1, \cdots, N - 1$. In the last step we have approximated $K = K(\rho, \rho \mathbf{V}, \rho_1, \cdots, \rho_{N-1}, \mathbf{J}_1, \cdots, \mathbf{J}_{N-1})$ with $\tilde{K} = K(\rho, \rho \mathbf{V}, \rho_1, \cdots, \rho_{N-1}, 0, \cdots, 0)$, that is, the latter is evaluated at the equilibrium. Truncating
the expansion above, we arrive at the multicomponent diffusion law

\[ J_i = -\varepsilon \sum_{j=1}^{N-1} D_{ij}(u) \left( \frac{\nabla p_j}{\rho_j} - \frac{\nabla p_N}{\rho_N} \right) \]  

(4.5)

with \( D_{ij}(u) \equiv \sum_{k,l=1}^{N-1} C_{ik} \tilde{K}_{kl} C_{lj} \) the multicomponent diffusion coefficients, which depend only on the conserved variable \( u \equiv (\rho, \rho V, \rho_1, \cdots, \rho_{N-1})^T \).

As to the relation (4.5), we have the following remarks.

**Remark 4.1**

- **The diffusion matrix**

\[ D(u) \equiv [D_{ij}(u)]_{(N-1)\times(N-1)} = [C_{ij}]_{(N-1)\times(N-1)} \tilde{K} [C_{ij}]_{(N-1)\times(N-1)} \]

is symmetric and positive definite, since \([C_{ij}]_{(N-1)\times(N-1)}\) is invertible, \( \tilde{K} \) is positive definite and they are both symmetric.

- According to the general theory (see, e.g., [16]), the term \( \left( \frac{\nabla p_j}{\rho_j} - \frac{\nabla p_N}{\rho_N} \right) \) can be expressed as the spatial gradient of the equilibrium-entropic force. To see this, we recall (3.2) that

\[ \eta(U) = \sum_{i=1}^{N} \left( \int_{z_i}^{\rho_i} \frac{p_i(z)}{z^2} dz + \frac{1}{2\rho_i} |\rho_i V_i|^2 \right) \]

\[ = \sum_{i=1}^{N} \left( \int_{z_i}^{\rho_i} \frac{p_i(z)}{z^2} dz + \frac{1}{2\rho_i} |\rho_i V + J_i|^2 \right). \]

Thus, we have

\[ \eta^q(u) \equiv \eta(U)|_{J_i=0} = \sum_{i=1}^{N} \rho_i \int_{z_i}^{\rho_i} \frac{p_i(z)}{z^2} dz + \frac{\rho_i}{2} |V|^2. \]

For fixed \( \rho \) and \( V \), we compute

\[ \nabla \frac{\partial \eta^q(u)}{\partial \rho_i} = \frac{\nabla p_i}{\rho_i} \frac{\nabla p_N}{\rho_N}, \quad i = 1, 2, \cdots, N-1. \]

Consequently, the multicomponent diffusion law (4.5) can be rewritten as

\[ J_i = -\varepsilon \sum_{j=1}^{N-1} D_{ij}(u) \nabla \frac{\partial \eta^q(u)}{\partial \rho_j}. \]

Namely, the thermodynamic fluxes \( J_i \) are expressed in term of the equilibrium-entropic (thermodynamic) forces \( \nabla \frac{\partial \eta^q(u)}{\partial \rho_j} \), with the coefficient matrix \( D(u) \) symmetric. This looks like the Onsager reciprocal relation [19], while the definition of the thermodynamic forces is consistent to those in the literature [12, 13].

With the diffusion law (4.5), the system (4.1)–(4.3) can be approximated formally by the following
second-order partial differential equations

\[
\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{V}) = 0,
\]
\[
\frac{\partial (\rho \mathbf{V})}{\partial t} + \nabla \cdot (\rho \mathbf{V} \otimes \mathbf{V}) + \nabla \sum_{j=1}^{N} p_j = 0,
\]
\[
\frac{\partial \rho_i}{\partial t} + \nabla \cdot (\rho_i \mathbf{V}) = \varepsilon \nabla \cdot \left( \sum_{j=1}^{N-1} D_{ij}(\mathbf{u})(\frac{\nabla p_j}{\rho_j} - \frac{\nabla p_N}{\rho_N}) \right)
\]

for \(i = 1, \ldots, N-1\). Thanks to the remark above, the last equations can be written as

\[
u_t + \nabla \cdot G(u) = \varepsilon \nabla \cdot \left( \sum_{k=1}^{d} \left( D(u)(\frac{\partial^2 \eta_{uu}^{eq}(u)}{\partial \mathbf{v}})_{x_k} \right) z_k \right) \tag{4.6}
\]

with \(G(u) = (\rho \mathbf{V}, \rho \mathbf{V} \otimes \mathbf{V} + \sum_{j=1}^{N} p_j I_d, \rho_1 \mathbf{V}, \ldots, \rho_{N-1} \mathbf{V})^T\) and \(\nu = (\rho_1, \ldots, \rho_{N-1})^T\). This is the second-order partial differential equations (2.5) in [16] with

\[
B^{jk}(u) = \delta_{jk}\text{diag}(0_{d+1}, D(u))\eta_{uu}^{eq}(u),
\]

where \(0_{d+1}\) is the zero-matrix of order \((d+1)\). It is known from [16] that \(\eta^{eq}(u)\) is strictly convex and therefore its Hessian \(\eta_{uu}^{eq}(u)\) is positive definite. Moreover, it was showed in [16] that the system (4.6) has a nice entropy structure, which is extremely important mathematically and numerically.

With the above expression of \(B^{jk}(u)\), we can simply show the following proposition.

**Proposition 4.1** The system (3.1) satisfies the isotropy condition (*) of Theorem 2.1 in [16].

**Proof.** By Lemma 3.2 in [16], it suffices to show that the null-space of the symbol matrix

\[
B(u, \xi) \equiv \sum_{j,k=1}^{d} B^{jk}(u)(\eta_{uu}^{eq}(u))^{-1} \xi_j \xi_k = \sum_{j,k=1}^{d} \delta_{jk}\text{diag}(0_{d+1}, D(u))\xi_j \xi_k = \text{diag}(0_{d+1}, D(u))|\xi|^2
\]

is independent of \(u\) and \(\xi \in \mathbb{R}^d\setminus\{0\}\). This is clear thanks to the positive definiteness of \(D(u)\).

Thanks to this proposition and the conservation-dissipative structure, we use Theorem 2.1 in [16] and get the following conclusion.

**Theorem 4.1** Under the conditions of Theorem 3.1, let \(s > d/2 + 1\) be an integer. Assume that \(\bar{U}(\varepsilon)\) as initial data for PDEs (3.1) and \(\bar{u}(\varepsilon)\) for (4.6) are in \(H^s(\mathbb{R}^d)\) for \(\varepsilon > 0\), satisfy

\[
\|\tilde{u}(:, \varepsilon) - \bar{u}(\cdot, \varepsilon)\|_s = O(\varepsilon^2)
\]

with \(\tilde{u}(x,\varepsilon)\) the conserved mode of \(\tilde{U}(x,\varepsilon)\), and all the components corresponding to densities have positive lower bounds. Then there exist \(\varepsilon\)-independent positive constants \(T_\ast > 0\) and \(K(T_\ast)\) such that the
solutions to PDEs (3.1) and (4.6) with the above initial data, denoted by $U^\varepsilon(x,t)$ and $u(x,t)$, are in $C([0,T_\ast], H^s(R^d))$ and

$$\sup_{t \in [0,T_\ast]} \| u^\varepsilon(\cdot,t) - u_\varepsilon(\cdot,t) \|_s \leq K(T_\ast)\varepsilon^2,$$

where $u^\varepsilon(x,t) \equiv (\rho^\varepsilon, \rho^\varepsilon \nabla^\varepsilon, \rho_1^\varepsilon, \ldots, \rho_{N-1}^\varepsilon)$ is the conserved mode of $U^\varepsilon(x,t)$.

The notation used in Theorem 4.1 is standard: For a nonnegative integer $s$, $H^s(R^d)$ is the space of functions whose distribution derivatives of order $\leq s$ are all in $L^2$ and we use $\| U \|_s$ to denote the standard norm of $U \in H^s$. $C([0,T], X)$ represents the space of continuous functions on $[0,T]$ with values in a Banach space $X$.

Finally, we give a detailed comparison with the multicomponent diffusion law derived in [9].

**Remark 4.2** In [9], Lam introduced $N$ numbers $\omega_i$ satisfying $\sum_{i=1}^N \omega_i \neq 0$ and replaced the collision coefficients $K_{ij}$ with $\hat{K}_{ij} \equiv K_{ij} + \omega_i \rho_j$ based on the zero-net-flux condition. By using the invertibility of the rank-one modification $[\hat{K}_{ij}]$ of the singular matrix $[K_{ij}]$, he derived the following multicomponent diffusion law

$$J_i = -\varepsilon \sum_{j=1}^N \rho_i \hat{D}_{ij} d_j, \quad i = 1, 2, \ldots, N,$$

where $[\hat{D}_{ij}] = p[\hat{K}_{ij}]^{-1}$ and $d_j = \nabla \rho_j + (\rho_j p_0 - \rho_i p_j) \nabla (\ln p)$ with $p = \sum_{i=1}^N p_i$. It is not difficult to verify the following relation

$$p \frac{\nabla d_j}{\rho_j} - \nabla \rho_j \frac{\nabla p_j}{\rho_j} = \nabla p_i - \nabla p_N \frac{\nabla p_j}{\rho_j}$$

between our thermodynamic forces and Lam’s. Obviously, Lam’s multicomponent diffusion law is not completely determined in general, for the diffusion matrix $[\hat{D}_{ij}]$ depends on the arbitrary parameters $\omega_i$. Moreover, it is not clear whether $[\hat{D}_{ij}]$ is symmetric or positive-definite, while so is ours $[D_{ij}(u)]$.

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**References**

[1] W. H. Furry, *On the elementary explanation of diffusion phenomena in gases*, Am. J. Phys. 16 (1948), 63.

[2] C. F. Curtiss and J. O. Hirschfelder, *Transport properties of multi-component gas mixtures*. J. Chem. Phys. 17:6 (1949), 550–555.

[3] F. A. Williams, *Elementary derivation of the multicomponent diffusion equation*, Am. J. Phys. 26 (1958), 467.
