Sustainable resistance switching performance from composite-type ReRAM device based on carbon Nanotube@Titania core–shell wires

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A novel nanocomposite-based non-volatile resistance switching random access memory device introducing single-walled carbon nanotube (SWCNT)@TiO2 core–shell wires was proposed for flexible electronics. The SWCNT was de-bundled by ultrasonication with sodium dodecylbenzene sulfonate (SDBS), and then the TiO2 skin layer on the SWCNT surface was successfully introduced by adding benzyl alcohol as a weak surfactant. The nanocomposite resistance switching layer was composed of the SWCNT@TiO2 core–shell wires and poly(vinyl alcohol) (PVA) matrix by a simple spin-coating method. The device exhibited reproducible resistance switching performance with a remarkably narrow distribution of operating parameters (VSET and VRESET were 2.63 ± 0.16 and 0.95 ± 0.11 V, respectively) with a large RON/ROFF ratio of 105 for 200 consecutive switching cycles. Furthermore, the excellent resistance switching behavior in our device was maintained against mechanical stress up to 105 bending test. We believe that the nanocomposite memory device with SWCNT@TiO2 core–shell wires would be a critical asset to realize practical application for a flexible non-volatile memory field.

Increasing data storage, which demands computing and imaging electronics, has been necessitating the development of high-performance memory devices with fast operation speed, high data storage, and low power consumption1–4. Moreover, ubiquitous platforms with flexible computing systems have become a critical asset for futuristic technology5,6. Therefore, it is crucial to achieve excellent and sustainable electrical and mechanical performances in the field of storing information7. Among several next-generation memory candidates, resistance switching random access memory (ReRAM) device, which is based on resistance switching phenomenon by two distinguishable resistance states, has attracted much attention due to high operation speed, non-volatility, simple structure, low-power consumption, and compatibility to conventional Si-based fabrication process1,2,8–10.

The flexibility has been aroused as a new requirement for the next-generation semiconductor devices due to demonstrating the wearable application. For a flexible application of the ReRAM device, the transition metal oxide (TMO)-based ReRAM devices are widely used due to their excellent resistance switching performance. However, the TMOs have a fatal weakness in flexible applications due to their lack of flexibility. Organic-based ReRAM devices, in contrast, have good flexibility but low yields and poor switching performance. The nanocomposite structures composed of a combination between inorganic clusters as a resistance switchable filler (RSF) and a polymer/organic matrix have currently emerged as the promising components for next-generation flexible ReRAM memory devices because the nanocomposite system can ideally take excellent resistance switching properties of the inorganic material and good mechanical and optical properties of the organic-based material11–15.

Carbon-based materials such as graphenes, graphite nanosheets, carbon nanofibers, and carbon nanotube (CNT) have recently attracted much attention owing to their fascinating properties such as a tunable bandgap,
high electron mobility, and quantum electronic transport\(^{16,17}\). Among them, CNT is a promising material for the flexible electronics due to superior electrical and mechanical properties. Moreover, a large aspect ratio of the CNT has an advantage of electrical percolation to connect between the bottom and the top electrodes of the ReRAM device. \textit{Hwang} et al. reported the nanocomposite-based ReRAM device introducing B- and N-doped CNTs dispersed in a polystyrene matrix\(^{18}\). \textit{Chaudhary} et al. also reported the non-volatile resistive switching effect by a poly(3-hexylthiophene)-CNT composite films\(^{19}\). \textit{Zhao} et al. reported rewritable composite-type memory based on polymethyl methacrylate-CNT layer\(^{20}\), and polyurethane-CNTs composite layer was introduced as a resistive layer by Li group\(^{21}\). Although the resistive devices based on charge trap- or filamentary-controlled TiO\(_2\) layer on the SWCNT surface in this study.

The de-bundling process of SWCNTs is essential due to its \(\pi-\pi\) interaction and van der Waals interaction between them\(^{22}\). Chemical treatment with strong acids, such as H\(\text{NO}_3\) or H\(\text{SO}_4\), is widely used for the de-bundling of the CNTs. The strong oxidizing agents preferentially disrupt the aromatic ring structure of the CNTs\(^{23}\). Some papers have reported that the disrupted structure of the carbon-based materials can be changed by externally applied bias, being able to induce resistive switching phenomenon\(^{24-27}\). In our system, we should avoid the resistive change of the CNT because our concept achieved the resistive switching behavior by employing the TiO\(_2\) resistive skin layer on the SWCNT surface. The role of SWCNT is to transport the injected electrons from one electrode to another when the conducting filament is formed in the TiO\(_2\) resistive switchable layer. Therefore, the de-bundling process with the physical method was employed instead of conventional chemical treatment with strong acid. As shown in Fig. 1a and b, we confirmed that SWCNTs were successfully de-bundled by the physical de-bundling route. The quality of the resulting SWCNTs could be confirmed by the Raman spectroscopy, as shown in Fig. 1c. Raman spectroscopy is an effective method to characterize the bonding detail of carbon-based materials, such as the bonding type, domain size, and internal stress. In the broad Raman peaks, D-band (1322 cm\(^{-1}\)) is associated with the \(\text{sp}^2\) bonding structure, and G-band (1590 cm\(^{-1}\)) indicates the \(\text{sp}^2\) bonding structure of the resulting SWCNTs. When the ratio of the intensity of the D and G bands (\(I_D/I_G\)) before and after the physical separation process was calculated, the \(I_D/I_G\) ratio of resulting SWCNTs before and after the de-bundling process is hardly changed from 0.07 to 0.11, indicating that the resulting SWCNTs were successfully separated without developing defect on its surface\(^{23}\).

To introduce the TiO\(_2\) resistive layer on the SWCNT, the surface characteristics of the SWCNT must be changed because the inherent property of the SWCNT is hydrophobic\(^{28}\). When the TiO\(_2\) is synthesized by the sol–gel method, it is hard to adsorb the TiO\(_2\) sol on the SWCNT surface\(^{29}\). Therefore, the SWCNT surface should be functionalized to provide an adsorption site with the TiO\(_2\) sol. The common approach is to develop covalent bonding on the SWCNT surface by ultrasonication in strong acid. However, already mentioned above, we have to avoid the destruction of the aromatic ring of SWCNT. Moreover, the acid treatment allows little control over the location, nature, and quantity of the introduced functional groups, leading to a non-uniform coating of the TiO\(_2\) layer on the SWCNT surface, as shown in Fig. 2a. Thus, BA is considered for introducing a high-quality TiO\(_2\) layer on the SWCNT surface in this study.

**Results and discussion**

The de-bundling process of SWCNTs is essential due to its \(\pi-\pi\) interaction and van der Waals interaction between them\(^{22}\). Chemical treatment with strong acids, such as H\(\text{NO}_3\) or H\(\text{SO}_4\), is widely used for the de-bundling of the CNTs. The strong oxidizing agents preferentially disrupt the aromatic ring structure of the CNTs\(^{23}\). Some papers have reported that the disrupted structure of the carbon-based materials can be changed by externally applied bias, being able to induce resistive switching phenomenon\(^{24-27}\). In our system, we should avoid the resistive change of the CNT because our concept achieved the resistive switching behavior by employing the TiO\(_2\) resistive skin layer on the SWCNT surface. The role of SWCNT is to transport the injected electrons from one electrode to another when the conducting filament is formed in the TiO\(_2\) resistive switchable layer. Therefore, the de-bundling process with the physical method was employed instead of conventional chemical treatment with strong acid. As shown in Fig. 1a and b, we confirmed that SWCNTs were successfully de-bundled by the physical de-bundling route. The quality of the resulting SWCNTs could be confirmed by the Raman spectroscopy, as shown in Fig. 1c. Raman spectroscopy is an effective method to characterize the bonding detail of carbon-based materials, such as the bonding type, domain size, and internal stress. In the broad Raman peaks, D-band (1322 cm\(^{-1}\)) is associated with the sp\(^2\) bonding structure, and G-band (1590 cm\(^{-1}\)) indicates the sp\(^2\) bonding structure of the resulting SWCNTs. When the ratio of the intensity of the D and G bands (\(I_D/I_G\)) before and after the physical separation process was calculated, the \(I_D/I_G\) ratio of resulting SWCNTs before and after the de-bundling process is hardly changed from 0.07 to 0.11, indicating that the resulting SWCNTs were successfully separated without developing defect on its surface\(^{23}\).

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We investigated the effect of BA in the TiO$_2$ coating on the SWCNT surface. The molar ratio of BA for a reaction mixture was controlled under the condition of fixing the CNT concentration of 10 wt% of the total weight. The molar ratio of BA (m$_{BA}$) was Ti precursor:BA:EtOH:H$_2$O = 0.25:x:30:5, where x was varied with 10 (m$_{BA,10}$) and 30 (m$_{BA,30}$). In the morphological analysis by scanning electron microscopy (SEM) measurement, we could confirm that the introducing BA strongly affects the quality of introducing the TiO$_2$ layer on the surface of SWCNTs (Fig. 2c and d), compared with the resulting particles prepared without introducing BA (Fig. 2b). In addition, it was observed that the resulting SWCNTs without BA treatment were reaggregated after the sol–gel reaction. It is because BA is adsorbed on the SWCNT surface by π–π interactions between the aromatic SWCNT surface and the benzyl ring of BA$^{29,30}$. Moreover, the hydroxyl group of the BA provides hydrophilic property on the SWCNT surface, leading the uniform hydrolysis with the titanium precursor$^{31}$. The amount of BA in the reaction mixture also affects the quality of the TiO$_2$ coating significantly. According to the increasing molar ratio of BA in the mixture, the TiO$_2$ layer tended to coat uniformly on the entire surface of SWCNTs. It is because the high concentration of BA can provide the functionalized sites on the entire surface of SWCNTs. The thickness of the TiO$_2$ layer on SWCNT revealed about 15 nm, as shown in the inset image of Fig. 2d. Finally, as shown in Fig. 2e, while the ST-CSW before heat treatment was mostly amorphous nature, the TiO$_2$ layer exhibited the crystal structure of the anatase phase after heat treatment at 500 °C.

We prepared the composite-based ReRAM device with the 300 nm-thick composite layers embedding 1 wt% of ST-CSWs in PVA polymer supporter (Supplementary Figure S1). Then, the $I$–$V$ measurements were conducted for the device. The electroforming behavior was observed at 5.25 V, as shown in Fig. 3a. The forming process exhibited a two-step forming characteristic because the actual structure of the device has two metal–insulator–metal structures of Pt top electrode (TE)/TiO$_2$/SWCNT and SWCNT/TiO$_2$/Pt bottom electrode (BE). When the subsequent $I$–$V$ sweep measurements were conducted up to 200 cycles, the resistance switching behavior showed the typical unipolar resistance switching (URS) behavior. The SET [i.e. resistance switching from a high resistance state (HRS) to a low resistance state (LRS)] and RESET behaviors (switching back from LRS to HRS) were operated successfully without any failure or degradation of the operation parameters ($SET/RESET$ voltages and $R_{ON}/R_{OFF}$ ratio). The average values of the SET and RESET voltages were 2.63 ± 0.16 and 0.95 ± 0.11 V, respectively, demonstrating remarkably uniform switching operation within narrow voltage ranges. The distribution of the SET and RESET voltages, and LRS and HRS values for repeated 200 $I$–$V$ cycles were summarized in Fig. 3b and 3c, respectively. As shown in Fig. 3d, the composite-based ReRAM device exhibited the reproducible switching behavior in a pulse switching test up to ~ 5,000 switching cycles. The $R_{ON}/R_{OFF}$ ratio during the test was about 10$^4$. In addition, the resistance values and their standard deviations of the LRS and HRS were (3.48 ± 0.07) × 10$^4$ and (8.1 ± 4.9) × 10$^8$ Ω, respectively. This result obviously proved the stable switching operation of our memory device. When a retention test on the device was conducted at 358 K, the two resistance values were stable up to 10$^6$ s (Fig. 3e). Furthermore, cell-to-cell tests for 25 cells were performed. Figure 4 and Supplementary Figure S3 show $I$–$V$ curves and SET and RESET voltage values collected at the 100th sweep cycle of each cell, respectively, exhibiting very uniform switching characteristics with narrow distribution (SET and RESET voltage of 2.69 ± 0.17 and 0.92 ± 0.13 V, respectively). This remarkable resistance switching operation is due to a confined electric field effect.
derived from unique structural contact between one-dimensional ST-CSW and two-dimensional electrode. It is well-known that manipulating electric field in resistive switching memory device can control the nucleation and growth of conductive filament, achieving reliable resistive switching performance. In addition, when we prepared the composite-based ReRAM device including 3 and 5 wt% of ST-CSWs, the resistance switching characteristics including pristine current level and forming voltage was similar to that of the composite-based ReRAM device with 1 wt% of ST-CSWs (Supplementary Figure S2). This result means that the 1 wt% of ST-CSWs in the PVA matrix is sufficient to connect between the top and bottom electrodes.

To demonstrate the effect of the ST-CSW on the meaningful resistance switching results, the neat PVA film, the composite film embedded 10 wt% of TiO₂ particles prepared by the same sol–gel method with preparing ST-CSW, and the composite film with 1 wt% of SWCNT were examined (Fig. 5). The neat PVA film and the composite film introducing TiO₂ particles showed the typical $I-V$ curve of insulator when the voltages were applied up to 6 V. The composite film with 1 wt% of SWCNT showed the typical $I-V$ curve of the conductor.

Figure 3. (a) Resistance switching behavior of the composite ReRAM embedded 1 wt% of the ST-CSWs. (b) Cumulative probability graphs of the SET and RESET voltages of the composite ReRAM device. (c) Endurance of the composite ReRAM device. (d) Result of the pulse switching test for the composite-based memory device with 1 wt% of ST-CSWs for 5,000 cycles. The SET and RESET pulse heights [duration] of 3.75 V [0.25 μs] and 1.65 V [0.25 μs], respectively. (e) Retention result of the composite ReRAM device at 358 K.

Figure 4. (a) $I-V$ characteristics on 25 composite-based ReRAM cells with 1 wt% of ST-CSWs. The SET and RESET curves were collected at the 100th sweep cycle of each cell, respectively. (b) Cumulative probability of SET and RESET operating voltages on the cell-to-cell test.
Therefore, it can be concluded that the resistance switching behavior of the composite-based ReRAM device is originated from the CT-CSW.

To identify the switching mechanism of the device, the area-dependent resistance switching behavior was examined because TiO₂ is well-known as a material capable of both resistance switching mechanisms by the charge trapping or filament formation. The HRS and LRS values were independent of the varied TE sizes (Fig. 6a), implying that the resistance switching behavior of the device is originated from the filamentary conduction switching mechanism. To clarify this behavior, the temperature coefficient of resistance (TCR) test on the device was conducted with a temperature range from 303 to 343 K, based on the following equation:

\[ R_T = R_0 [1 + \alpha (T - T_0)] \]  

where \( \alpha \) is the TCR value. \( R_0 \) and \( R_T \) are the resistance of the LRS at the initial temperature \( T_0 \) and the resistance of the LRS at temperature \( T \), respectively. The calculated \( \alpha \) value from Fig. 6b of the device was \( 5.09 \times 10^{-5} \text{ K}^{-1} \). The value is comparable with the reported value of the Magnéli phase \( (1.47 \times 10^{-5} \text{ K}^{-1})^{34,35} \), implying that the resistance switching behavior of the device would be attributed to the formation of the conductive filament consisting of Magnéli phase by the electrical stimulus.

On the basis of the above results, the proposed resistance switching mechanism is illustrated in Fig. 7. When the positive bias is applied to the Pt TE, the resistance of the TiO₂ junction between the Pt TE and the SWCNT drops due to the phase transition from TiO₂ to Ti₄O₇ in the TiO₂ skin layer⁴⁶. And then, the second filament will be subsequently formed in the same manner at the TiO₂ junction between the SWCNT and Pt BE. The resistance state can be switched back LRS to HRS by a rupture of a formed conductive filament when a negative bias is applied to the Pt TE by joule heating effect derived from a higher current than the set compliance current.

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**Figure 5.** I–V curves of PVA film, PVA with 1 wt% TiO₂ particles prepared by sol–gel process, and PVA with 1 wt% SWCNT. (inset) magnified I–V curves of PVA film, and PVA with 1 wt% TiO₂ particles.

**Figure 6.** (a) Electrode-area dependent behaviors of the composite ReRAM device. (b) Temperature coefficient of resistance graph of the composite ReRAM device.
Finally, the feasibility of the composite-based ReRAM as a flexible memory device was examined, as shown in Fig. 8. For the flexible test, the composite film with ST-CSWs of 1 wt% was cast on poly(ethylene terephthalate) (PET) substrate with crossbar-patterned TE and BE (Supplementary Figure S4). During the flexibility examination, the device was deformed from a flat state to a bending radius of 2 mm, and the test was conducted up to $10^5$ bending cycles. The operating parameters (SET and RESET voltages and $R_{ON}/R_{OFF}$ ratio values) were maintained without any degradation. This remarkable stability against mechanical stress is related to the inherent elastic properties of the PVA matrix and the one-dimensional structure of the ST-CSW. Moreover, the physical interaction such as a hydrogen bonding between the TiO$_2$ surface of the ST-CSW and the PVA matrix can be formed.

**Conclusions**

SWCNT@TiO$_2$ core–shell wire was prepared as a novel resistance switchable filler in the nanocomposite resistance switching layer. The TiO$_2$ skin layer on the surface of the SWCNT was uniformly formed by π–π interactions between the aromatic SWCNT surface and benzyl ring of BA, leading the uniform hydrolysis with the titanium precursor. And then, its resistance switching performance was investigated against mechanical and electrical noise. The nanocomposite resistance switchable devices showed notable resistance switching behavior with remarkably narrow operation parameters with a large $R_{ON}/R_{OFF}$ ratio. The reliable operation of our device was
also sustained even in cell-to-cell tests for 25 cells. In addition, the notable resistance switching behavior with narrow operation parameters and the large $R_{ON}/R_{OFF}$ ratio was maintained against 10$^5$ bending cycles without any deterioration and failure. Based on the above results, we conclude that the nanocomposite resistance switching layer based on one-dimensional SWCNT@TiO$_2$ core–shell wire would be a promising platform for flexible electronics as an essential data storage component.

**Methods**

0.5 wt% of Sodium dodecylbenzene sulfonate (SDBS) and 1 wt% of SWCNT (E.C. 2.0, Meijo Nano carbon Co.) was added in 50 mL of EtOH, and then the solution was tip-sonicated for 1 h (750 W and 40% amplitude condition). Benzyl alcohol (BA) and deionized water were added and mixed by magnetic stirring at 0 °C for 12 h. The tetraethyl-orthotitanate (TBOT) was dissolved in ethanol and slowly dropped into the suspension. The final molar ratio of the mixture of TBOT:BA:EtOH:H$_2$O was 0.25:50:5:5 ($x = 10$ and 30). After one hour reaction, the precipitates were purified by vacuum filtering with EtOH washing. The resulting particles were freeze-dried for 36 h. Finally, the particles were heat-treated (150 °C for 1 h followed by 400 °C for 1 h) in a tube furnace for the crystallization of the TiO$_2$ skin layer. Pt bottom electrode (50-nm-thick) deposited on TiO$_2$/SiO$_2$/Si substrate of 1.25 × 1.25 cm was cleaned with acetone, deionized water, and then ethanol in an ultrasonic bath for 15 min, respectively. 1 wt% of ST-CSWs were mixed with 10 wt% Polyvinyl alcohol (PVA) solution ($M_w = 50,000$ g/mol) in deionized water. And then, the mixture was stirred at 70 °C for 6 h. The solution containing ST-CSWs was spin-cast on the Pt BE/TiO$_2$/SiO$_2$/Si substrate, followed by solvent removal at 70 °C for 24 h. Finally, 50 nm-thick Pt top electrode (TE) with a dot shape was deposited via the sputtering method with a metal shadow mask whose hole diameters were varied of 100, 200, 250, and 500 μm, respectively. The morphological analyses of the pristine SWCNT, the ST-CSW, and the composite film were conducted by scanning electron microscopy (SEM, JSM-6701F, JEOL). The crystal structure of the ST-CSW was characterized by X-ray diffraction (XRD, Dmax 2500). The current–voltage ($I–V$) measurement was performed using an Agilent 4156C analyzer under ambient conditions. The pulse measurement was conducted using a semiconductor parameter analyzer, Keithley 4200-SCS, with Keithley 4225-PMU (pulse measurement unit).

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**Author contributions**

Y.K.: Methodology, Investigation, Resources, Writing—original draft. M.K.: Investigation, Resources. J.H.H.: Investigation, Resources. T.W.K.: Methodology, Writing—review and editing. S.-S.L.: Conceptualization, Methodology, Supervision, Writing—review and editing, Funding acquisition. W.J.: Conceptualization, Methodology, Supervision, Writing—review and editing, Funding acquisition.

**Competing interests**

The authors declare no competing interests.

**Additional information**

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