We thank Referee #1 for the comments and suggestions. We have addressed every comment and made significant changes to the paper to improve the paper. Again, the referee’s comments are greatly appreciated.

**Referee Comments in black bold.**

**Authors’ Response in blue.**

**Changes in manuscript in Red italic.**

**Major comments**

This paper develops an automatic linear calibration method to calibrate the mass spectra for individual particles measured by newly developed HR-SPAMS. The method improves the current accuracy of mass-to-charge (m/z) measurement for single aerosol particles, based on the testing of laboratory-generated sea spray aerosol and atmospheric ambient aerosol. The authors provided the time series of peaks with small m/z differences and a comparison of particle classification between LR-SPAMS and HR-SPAMS. While this method may be applicable to the scientific community, there are still some limitations. The main criticism is the limited discussion of the analysis of newly generated mass spectra by the HR-SPAMS, and the lack of discussion on why only sea spray aerosols were selected. It is certainly TRUE that applying high-resolution data with enhanced mass calibration can significantly affect particle classification (identification). However, it is more important if there is new information obtained from the classification.

Response:

We agreed that the discussions of the analysis of newly generated mass spectra by the HR-SPAMS are limited and the new information obtained from the classification is not enough in the original manuscript. In the revised manuscript, we have added the more detailed discussions of the HR-SPAMS mass spectra and illustrated why we selected sea spray aerosols as a case study, which considers the sea spray aerosol chemical compositions are relatively simple, compared to the ambient aerosol chemical compositions. The calibration method was evaluated to apply both in simple and more complicated systems. Meanwhile, based on the referee’s comments, some new information, obtained from the classification, is added.

Changes in manuscript:

*Line 99-107:*

“In this study, we report a calibration method for single particle high resolution mass spectra data. Based on the assumption that the sea spray aerosol has relatively simple chemical composition while the ambient aerosol has more complex chemical composition, the performance of the calibration method had been evaluated in detail for these two aerosol systems with different complexity. In addition, the impact of using high resolution SPAMS data on particle classification by ART-2a algorithm was assessed. An open source code specific for
HR-SPAMS was made and we proposed the principle of this calibration method can be applied into some similar instruments, such as single particle mode Aerosol Mass Spectrometer (AMS).”

Line 273-280:
“More importantly, 139[C_2H_3O_5S^+] (the theoretical m/z value: -138.97) can be clearly distinguished from other possible assignments, such as 139[C_{12}H_7^+] with the m/z value of -139.55 and 139[AsO_4^3-] with the m/z value of -138.90. Moreover, 153[C_2H_3O_5S^+] with theoretical m/z value of -152.986 can be distinguished from other possible assignments, such as 153[C_{12}H_9^+] with m/z value of -153.070 and 153[Na_2Cl_3] with m/z value of -152.883. These two important organic ion peaks have been suggested to be the characteristic ion peaks for the organosulfates in secondary organic aerosols (Surratt et al., 2010; Surratt et al., 2007).”

Line 319-347:
“The ART-2a classification of the HR-SPAMS results (Fig.S6) show that the signal at 23[Na^+] in Type 2HR was stronger than Type 1HR while the signals at 26[CN] and 42[CNO] were weaker in Type1 HR. Meanwhile the averaged mass spectra of the Type 2HR showed the presence of 206[Pb^2+], 207[Pb]+ and 208[Pb]+, which are known to be harmful to human health (Das et al., 2018; Peng et al., 2020). Furthermore, particles of Type 2HR containing abundant secondary inorganic components like [NO_2^+], [NO_3^-] and [SO_4^{2-}], which originated from the aerosol aging processes (Dall’Osto and Harrison, 2012; Ma et al., 2016). In contrast, these two first particle types were lumped together into Type 1LR in the LR-SPAMS classification results (Fig.S5). Due to the merge of these two particle types, [Pb+] ions were not significant. Meanwhile, Type 3, 5HR classification results contain strong signals at 26[CN], 42[CNO], 46[NO_2], 62[NO_3] and 97[HSO_4], suggesting that these three types were from biomass burning or residential cooking burning. [K^+] is also another feature of this type particle emission (Bi et al., 2011; Hudson et al., 2004). There were obvious relative ion intensity differences at 26[CN], 42[CNO], 46[NO_2] and 62[NO_3] among these types, which implied that these three particle types might be from different burning sources or experienced different levels of aging (Luo et al., 2020). While these three particle types were lumped together as Type 2LR. This critical information which could be potentially used to distinguish particle sources and aging processes was lost. Additionally, Type 7HR can be assigned as ECOC type, based on its strong signals at [C^+]_4, [C_2H_2O_4^+] and [C,H,O_3^+]. Particles of this type may come from the primary emission sources, and the emitted black carbon particles would also form this type particles after absorbing some low volatile organic compounds in the atmosphere (Sodeman et al., 2005). 97[HSO_4]^- can be observed to have a weaker signal than 62[NO_3^-] and 46[NO_2^-], which implied that the secondary reaction of SO_2 as the precursor of HSO_4^- was not significant in the particle surface for ECOC type particles in this study (SULLIVAN and PRATHER, 2007). In contrast, the classification results of the LR-SPAMS were not so clear and generated less particle types. Given HR-SPAMS spectra have much more detailed chemical information about particles, we would propose that the ART-2a classification of HR-SPAMS might be more accurate.”

Specific comments:
1. Introduction: similar instruments, such as Aerosol mass spectrometer (AMS), also have
high-resolution versions. Is the calibration method identical to SPAMS? Inclusion of this in the introduction and discussion would be necessary for completeness.

We agree with the referee’s opinion, and decided to make an additional instruction in the introduction and conclusion for paper completeness. Our open source code of calibration method was specific for high-resolution SPAMS, while the principle of this calibration method can be applied into some similar instruments such as AMS.

Changes in manuscript:

Line 104-107:

“An open source code specific for HR-SPAMS was made and we proposed the principle of this calibration method can be applied into some similar instruments, such as single particle mode Aerosol Mass Spectrometer (AMS).”

Line 373-376:

“All the automatic linear calibration method codes specific for HR-SPAMS are open access and can be found at https://github.com/zhuxiaoqiang-fdu/zhuxiaoqiang-fdu. And we proposed the principle of this calibration method can be adopted in other aerosol mass spectrometers.”

2. Lines 80-85: In this section, it would be better to state the significance of why the calibration is required for each particle. While the authors noted that the ion peak position is still very susceptible to initial ion coordinate and speed, they did not provide details to show the significance.

We agree with the referee’s opinion, and added an additional statement of the reason why the calibration is required for each particle and provide the details to show the significance of the initial ion coordinate and speed.

Changes in manuscript:

Line 85-94:

“Unfortunately, in spite of resolution enhancement with this new technique, ion peak position was still very sensitive to initial ion coordinate and speed. Chudinov et al., has demonstrated that the ion peak shifts of 208[\text{Pb}^+] and 147[\text{Na(NO}_3)_2] could be varied in the range of ±10 ns and the ion start position could be varied in the range of ±150μm. As a result, substantial peak jittering is observed when switching between mass spectra of each individual particle. This peak jittering leads to a fact that isotopic pattern identification becomes more difficult by averaged mass spectrum(Chudinov et al., 2019). Furthermore, the peak jittering is different in each single particle mass spectrum. In other words, the calibration parameter for each mass spectrum should be significantly different and calibration is required for each particle.”

3. Line 192: “1,409 ambient particles were successfully calibrated”. Why some fraction of particles cannot be calibrated? I think the discussion of such an issue in section 3.3 should be moved here to provide clear reasoning. Such an obvious deficiency should also be stated in the abstract or conclusion.

We agree with the referee’s opinion that it’s an obvious deficiency. This is mainly due to the reason that we set the threshold of absolute ion intensity 15 a.u for marker ions in the
ambient aerosols and 8 a.u in the sea spray aerosols. And a particle was discarded from the spectrum database if it did not have enough reference ions (the minimum number of reference ions was set to be 5) in either positive or negative mass spectrum. And now we have made a clear reasoning in the section 3.2 and also stated in the conclusion.

Changes in manuscript:

Line 210-213:
“And 4,624 sea spray particles and 1,409 ambient particles were successfully calibrated. As some fraction of particles had been filtered because their mass spectrum did not have 5 or more reference peaks to conduct the calibrations. To mitigate this problem, we proposed some adjustments in the next section.”

Line 370-373:
“There is a deficiency of this HR-SPAMS calibration method, which has been showed that some fraction of particles cannot be calibrated due to the presence of weak signals of the marker ions. It can be mitigated by applying some additional marker ions.”

4. What is the matrix size produced by the HR-SPAMS? Is there a limit for the ART-2a to classify the matrix of particle mass spectra? Such information should be included.
As there is 25,000 bins in either positive or negative single mass spectrum and we handled about 1,400 particles in this study. The matrix size is about $25,000 \times 2 \times 1400 = 7 \times 10^7$. There was no absolute limit for ART-2a as long as the database was not extremely large, like the experiment data lasting for several months. And we have made the following additional statement to include this necessary information in our revised manuscript.

Changes in manuscript:

Line 305-306:
“The previous ambient aerosol SPAMS dataset (1,400 particles) was used for the matrix size of the ART-2a is around $7 \times 10^7$. “

5. Section 4.4: What kind of new information is provided when new matrix is included in the classification? I think it would be interesting if there is new information after the classification of newly calibrated mass spectra.
It is the same comment as the referee’s major comments. We agree with the referee’s comment that the discussions of our new classification results are not quite sufficient in our original manuscripts. And now we have made a more comprehensive discussions, as we responses to the referee’s major comments above.

6. Some peak ions should be added to Fig S5 and Fig S6 for clearance.
We have added the necessary peak ions in the Fig S5 and Fig S6 for clearance in our revised supplement.

Changes in supplementary material:
Figure S5 Classification results of the LR-SPAMS by ART-2a
7. Conclusion: It would be better to include some atmospheric implications for the identification of additional peaks, in particular, organic peaks. Currently, the authors showed that more particle types can be obtained, but it might not be meaningful enough for the scientific community.
We agree with the referee’s comments and here is the revision:

Changes in manuscript:

*Line 358-370:*

“With this method, HR-SPAMS can also determine the time series of organic and inorganic peaks, whose m/z are very close to each other (e.g. 41K+ with the theoretical m/z value at 40.96182 and C_{6}H_{5}+ with the theoretical m/z value at 41.03913). Important organic ion peaks, such as tracer peaks for secondary organic matter like 139[C_{2}H_{3}O_{5}S] and 153[C_{5}H_{5}O_{5}S], can be identified. More importantly, our ART-2a classification from HR-SPAMS dataset clearly showed a particle type containing heavy metals like Pb+, which was obviously ignored in the ART-2a classification from the LR-SPAMS dataset. More particle types were generated by the ART-2a classification of HR-SPAMS data compared to that of LR-SPAMS data, as the original biomass burning particle type can be divided into three more detailed types based on the different signals of 26[CN], 42[CNO], 46[NO_{2}] and other organic species like C_{x}H_{y} and C_{x}H_{y}O_{z}, implying different aerosol aging processes or burning conditions. Such detailed information may be critical to study the aging processes and source appointment of atmospheric aerosols.”

8.Grammar check

*Line 70: LDI?
Line 77: “A SPAMS”
Line 99: “accessed”*

We have made correction to the grammar issues.

Changes in manuscript:

*Line 71-72:*

“However, SPAMS with laser desorption/ ionization (LDI) method has several serious limitations (Manuel et al., 2006; Wenzel et al., 2003).”

*Line 79:*

“A higher mass resolution version of the SPAMS with better m/z accuracy is needed.”

*Line 103-104:*

“In addition, the impact of using high resolution SPAMS data on particle classification by ART-2a algorithm was assessed.”
Authors’ response to referee comment RC2 on manuscript
“Development of an Automatic Linear Calibration Method for High Resolution Single Particle Mass Spectrometry: Improved Chemical Species Identification for Atmospheric Aerosols”

We thank Referee #2 for the comments and suggestions. We have addressed every comment and made significant changes to improve the paper. Again, the referee’s comments are greatly appreciated.

Referee Comments in black bold.
Authors’ Response in blue.
Changes in manuscript in Red italic.

Major comments
This study reports the development of an automatic linear calibration method for analyzing mass spectral data acquired with single particle mass spectrometers with mass resolution of ~ 2000. The paper also shows the successful application of this method to analyzing lab generated sea spray particles and some ambient aerosols. This work is important given the broad application of single particle mass spectrometry in atmospheric studies and aerosol research and improvement of the chemical resolution of this technique is important. The scope of the work fits well within AMT and the manuscript is generally well written. I recommend acceptance for publication after following comments are addressed.
Response:
We are grateful for the comments given by the referee and hope our new automatic linear calibration method can be applied into more versions of aerosol mass spectrometry for atmospheric studies. Also we have addressed all the comments in the following paragraphs.

Specific comments:

1. Line 119, change to “dried by”
We have made correction to that.
Changes in manuscript:
Line 135-136:
“The ambient particles were dried by a diffusional dryer before being sampled by the HR-SPAMS.”

2. Fig. S2, the caption for this figure needs to be rewritten to better present the information content.
We have rewritten the caption of the Fig.S2 to better present the information content.
Changes in supplementary material:
“Fig.S2 partial enlarged detail in the single particle mass spectra for the
explanation of the m/z bin value”

Changes in manuscript:

Line 141-144:

“Noticeably, due to the technical limitation of data acquisition, the whole HR-SPAMS spectrum is not continuous but divided by a large number of m/z bins, which are described in Fig.S2 (a partial enlarged detail in the single particle mass spectra) and can be viewed as the probability density histogram of the m/z.”

3. Line 171, what does a.u. stand for? How were the thresholds selected?

a.u. stands for the arbitrary unit, which is used widely in the averaged aerosol mass spectra for ion intensity. The absolute ion intensity threshold adopted in the YAADA (Yet Another ATOFMS DATA Analyzer, www.yaada.org) for SPAMS is 5 a.u., so the thresholds we used (15 a.u. for ambient aerosol and 8 a.u. for sea spray aerosol) can be considered reasonable just in case of the interfering signals in the aerosol mass spectra.

4. The description on Step 3 given in the paragraph on pages 6 and 7 is a bit hard to follow. How exactly is the calibration conducted? Are the measured m/z bins determined from the “traditional method” mentioned in Step 0? What exactly is the “traditional method” involved? How many bins are selected for each m/z?

(1) The detailed statement of the Step 3 is that: First, we will get the measured m/z bins determined from the “traditional method”, just as described in the Step 0. And these bins were not sufficiently accurate. Then, we picked up some reference ions to make a linear regression between the two set of variables (measured vs. theoretic reference ion m/z bin values). The obtained calibration parameters (a slope and an intersect) from linear regression were used to calibrate every bin value for this mass spectrum. However, the m/z bin values are fixed numbers (they are not continuous). So we had to assign the calibrated m/z value to its closest m/z bin value. Finally, the correct aerosol mass spectra can be acquired. We had made some revisions to make it easier to read in the Step3;

(2) The measured m/z bin values were determined from the “traditional method” mentioned in the Step 0;

(3) The “traditional method” was more like a coarsely-calibrated method, which usually selected a few particles with distinct ion patterns. Fig3 and Table S1&S2 had reported there were around five of larger bin numbers offset for the ionized species. What’s more important is that every particle mass spectra were different from each other and needed its own calibration parameters. So the “traditional method” is not enough for the SPAMS data processing.

(4) One specific bin was selected for each m/z as shown in the Fig.S2

Changes in manuscript:

Line 193-195:
“The measured m/z bin values of the reference ions mentioned in the Step0 were calibrated based on their theoretic (or true) m/z bin values.”

Line 197-199:
“Then we used these parameters to make the calibration for every bin value in this mass spectra. Finally, the m/z of the whole spectrum had been corrected.”

5. Give units for “measurement m/z” and “theoretical m/z” on the axis labels in all the Figures presented in this paper.
The unit is Dalton. We have updated all m/z axis labels in the revised paper.

6. Figure 3, the symbols are hard to differentiate, consider to revise. The spectra a and b look identical, are they really represent sea spray aerosol and ambient aerosol respectively?
Thanks for pointing this out. It was a mistake and has been corrected now. We have also separated the figure into 2 panels to make them easier to differentiate.
Changes in manuscript:
Line 548-553:
‘a.’

Figure 3. Probability distributions of the marker peak locations before and after Automatic Linear Calibration (AL-Cal) for (a) sea spray aerosol and (b) ambient aerosol”

7. Figure 4, what is “error limits”? How was is calculated?
The “error limit” is the delta bin number which is concluded from the Table S1&S2 and Fig.3,
representing the accepted error range (around 3 bin numbers) after the calibration for a m/z bin value.

Changes in manuscript:

**Line 252-254:**

“Figure 4 reports the average positive and negative mass spectra for the laboratory generated sea spray aerosols and the error limits mean the concluded accepted error range.”

7. **Line 250, change to “Ca2+”**

Actually the ion fragment generated from the SPAMS can only carry one charge, so the Ca$^+$ is reasonable.

9. **Line 266, define “LR-SPAMS”?**

We have given a definition for that.

Changes in manuscript:

**Line 295-297:**

“In contrast, it is impossible for a low resolution (LR)-SPAMS to provide such detailed time variation measurement of these peaks.”
Development of an Automatic Linear Calibration Method for High Resolution Single Particle Mass Spectrometry: Improved Chemical Species Identification for Atmospheric Aerosols

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Abstract

The mass resolution of laser desorption ionization (LDI) single particle aerosol mass spectrometry (SPAMS) is usually low (~500), which has been greatly improved by recent development of delayed ion extraction technique. However, due to large fluctuations among LDI processes during each laser shot, accurate calibration of mass-to-charge ratio for high resolution SPAMS spectra is challenging. Here we developed an automatic linear calibration method to improve the accuracy of mass-to-charge (m/z) measurement for single atmospheric aerosol particles. Laboratory generated sea spray aerosol and atmospheric ambient aerosol were tested. After the calibration, the fluctuation ranges of the reference ions (e.g. Pb\(^+\) and SO\(_4\)\(^+\)) m/z reaches ±0.018 for sea spray aerosol and ±0.024 for ambient aerosol in average mass spectra. With such m/z accuracy, the HR-SPAMS spectra of sea spray aerosol can easily identify elemental compositions of organic peaks, such as C\(_x\), C\(_x\)H\(_y\) and C\(_x\)H\(_y\)O\(_z\). While the chemical compositions of ambient aerosols are more complicated, C\(_x\)H\(_y\), C\(_x\)H\(_y\)O\(_z\) and CNO peaks can also be identified based on their accurate mass. With the improved resolution, the time series of peaks with small m/z differences can be separated and measured. In addition, it is also found that applying high resolution data with enhanced mass calibration can significantly affect particle classification (identification) using the ART-2a algorism, which classify particles based on similarities among single particle mass spectra.
1. Introduction

Atmospheric aerosols can significantly impact radiative forcing, cloud formation and human health (Ackerman et al., 2004; Zhang and Kin-Fai, 2012). They originate from various sources and undergo many atmospheric aging processes, resulting in an extremely complicated mixture of particles with a large range of sizes and chemical compositions. This mixture is usually referred as “mixing state”. Measurement of aerosol mixing state requires single particle characterization techniques. Utilizing laser ablation/ionization of single aerosol particle, Single Particle Aerosol Mass Spectrometer (SPAMS) has been widely used to measure chemical compositions, sizes and refractory index of aerosols in real-time (Moffet and Prather, 2009; Murphy, 2010; Sullivan and Prather, 2005). Based on this technique, ART-2a and other algorithms had been developed to classify the ambient particles based on their mass spectra and identify their sources (Reinard et al., 2007; Zelenyuk and Imre, 2009).

However, SPAMS with laser desorption/ionization (LDI) method has several serious limitations (Manuel et al., 2006; Wenzel et al., 2003). A major issue is that the mass resolution of the SPAMS is relatively low (~500) and the accuracy of m/z (mass to charge ratio) is usually at integer level, resulting in uncertainties about the identification of chemical species (Nash et al., 2006; Pratt and Prather, 2012; Qin et al., 2006). Due to the low mass resolution, many organic and inorganic peaks cannot be separated, such as K⁺/C₃H₃⁺ from the m/z peak of 39, Al⁺ C₂H₅⁺ from the m/z peak of 27 and CN⁻/C₂H₅⁻ from the m/z peak of -26 (Li et al., 2018). To better identify these particulate chemical species, a higher mass resolution version of the SPAMS with better m/z accuracy is needed.

Recently, Li et al., significantly increased SPAMS’s mass resolution to ~2000 by applying delayed ion extraction technique, which combined a standard rectangular extraction pulse with an exponential pulse (Li et al., 2018). This new SPAMS is called high resolution (HR)-SPAMS. Unfortunately, in spite of resolution enhancement with this new technique, ion
peak position was still very sensitive to initial ion coordinate and speed. Chudinov et al., has demonstrated that the ion peak shifts of $208[\text{Pb}^+]$ and $147[\text{Na(NO}_3)_2^-]$ could be varied in the range of $\pm 10\text{ ns}$ and the ion start position could be varied in the range of $\pm 150\mu\text{m}$. As a result, substantial peak jittering is observed when switching between mass spectra of each individual particle. This peak jittering leads to a fact that isotopic pattern identification becomes more difficult by averaged mass spectrum (Chudinov et al., 2019). Furthermore, the peak jittering is different in each single particle mass spectrum. In other words, the calibration parameter for each mass spectrum should be significantly different and calibration is required for each particle. Therefore, in order to get accurate m/z, Chudinov et al. used several peaks with known m/z to calibrate every SPAMS spectrum for Pb(NO$_3)_2$ and NaI particles produced from an atomizer.

In this study, we report a calibration method for single particle high resolution mass spectra data. Based on the assumption that the sea spray aerosol has relatively simple chemical composition while the ambient aerosol has more complex chemical composition, the performance of the calibration method had been evaluated in detail for these two aerosol systems with different complexity. In addition, the impact of using high resolution SPAMS data on particle classification by ART-2a algorithm was assessed. An open source code specific for HR-SPAMS was made and we proposed the principle of this calibration method can be applied into some similar instruments, such as single particle mode Aerosol Mass Spectrometer (AMS).

However, atmospheric particles are extremely complicated with a wide range of chemical compositions and sizes (Zhang et al., 2013), which brings much greater challenge to properly calibrating each SPAMS mass spectra and obtaining accurate m/z measurement. We need to develop a new MS calibration method for atmospheric aerosols and evaluate its performance comprehensively.
2. Experimental Section

2.1 High Resolution Single Particle Aerosol Mass Spectrometer (HR-SPAMS)

The detailed description of HR-SPAMS (Hexin Analytical Instrument Co., Ltd., China) can be found elsewhere (Li et al., 2018). Briefly, a HR-SPAMS consists of an aerodynamic lens as its particle inlet, two laser beams system for particle sizing, a UV laser for LDI and a bipolar time-of-flight mass analyzer for the detection of positive and negative ions. Positive and negative ions are detected by two z-shape bipolar TOF reflectron mass analyzers. The size detection range of HR-SPAMS is 200-2000 nm. As introduced before, this HR-SPAMS used delayed ion extraction technique to enhance its mass resolution.

2.2 Laboratory generated sea spray aerosol

Sea spray aerosol was produced by water jet method. In a sea spray aerosol production tank, a seawater jet was hitting seawater surface and producing bubbles, which would rise to the surface and burst. Bubble bursting process produces sea spray aerosols. Seawater was collected at Fengxian, Shanghai (30°92′N and 121°47′E) on March 30th (Fig.S1).

2.3 Ambient aerosol sampling

Ambient aerosol sampling was conducted at Fudan university, Shanghai (31°20′N and 121°30′E) on May 29th 2019 (Fig.S1). The ambient particles were dried by a diffusional dryer before being sampled by the HR-SPAMS.

3. Development of Calibration Methods

3.1 Automatic linear calibration method

To improve the accuracy of m/z for HR-SPAMS spectra, an automatic linear calibration method has been developed. Noticeably, due to the technical limitation of data acquisition, the whole HR-SPAMS spectrum is not continuous but divided by a large number of m/z bins, which are described in Fig.S2 (a partial enlarged detail in the single particle mass spectra) and can be viewed as the probability density histogram of the m/z. Here we denote “m/z bin value” as the median m/z value of each bin.

The linear calibration method is described as the following steps:
Step 0: The SPAMS data was coarsely-calibrated by the traditional method, which usually selected a few particles with distinct ion patterns, i.e. the molecular composition of some distinct peaks in the mass spectra can be easily identified. Then, the time of flight of these peaks and the true m/z of the corresponding ions were used to calculate a set of calibration parameters for positive and negative spectra. The parameters were finally applied to the whole mass spectra dataset, and coarsely-calibrated was completed.

Step 1: a pool of ion peaks in the single particle mass spectra were selected as the potential m/z calibration reference ions. The selection criteria are (1) these peaks should be present in most of the spectra; (2) the identification of these ion peaks should not be significantly affected by other adjacent peaks. For example, 27[Al]^+ was not selected, as its adjacent peak 27[C2H3] may affect the peak shape and identification of 27[Al].

According to the previous research, possible peak assignments for the m/z of reference ions for sea spray aerosol and ambient aerosol were listed on Table 1(Bertram et al., 2018; Collins et al., 2014; Tsunogai et al., 1972; Wang et al., 2016; Wang et al., 2019). For sea spray aerosol, according to several studies (Bertram et al., 2018; Collins et al., 2014; Tsunogai et al., 1972), the reference ions with m/z 23 24 39 -35 -37 were 23[Na]^+ 24[Mg]^+ 39[K]^+ -35[Cl]^- and -37[Cl]^- respectively. And Collins et al. shows that the reference ions with m/z 81, 83, -26, -42, -58, -129, and -131 were 81[Na2Cl]18 , 83[Na2Cl]18 , -26[CN] , -42[CNO]18 , -58[NaCl]18 , -129[MgCl3]18 , and -131[MgCl3]18 , respectively (Collins et al., 2014). Due to the fact that Na, Mg and K were abundant in sea spray aerosol, the reference ions with m/z 113 and 115 should be 113[K2Cl]18 and 115[K2Cl]18. Thus, in this study, we select 23[Na]^+ , 24[Mg]^+ , 39[K]^+ , 81[Na2Cl]18 , 83[Na2Cl]18 , 113[K2Cl]18 , 115[K2Cl]18 , -35[Cl]^- , -37[Cl]^- , -26[CN] , -42[CNO]18 , -129[MgCl3]18 , -131[MgCl3]18 , -58[NaCl]18 as the potential reference ions for sea spray aerosols.

For the ambient aerosol, according to the previous ambient SPAMS measurements(Wang et al., 2016; Wang et al., 2019), the reference ions with m/z 12, 23, 36, 39, 56, 207, 208, and 209 were assigned to 12[C]^+ , 23[Na]^+ , 36[C3]18 , 39[K]^+ , 56[Fe]^+ , 207[Pb]^+ , 208[Pb]^+
and 209\[Pb\]^+, the reference ions with m/z -26, -35, -46, -62, -96, and -97 were assigned to -26[CN]^-,-35[Cl]^-, -46[CNO]^-, -62[NO2]^-, -96[SO4]^-, and -97[HSO4]^-. So in this study, we select the 12[C]^+, 23[Na]^+, 39[K]^+, 36[C3]^+, 56[Fe]^+, 208[Pt]^+, 206[Pt]^+, 207[Pt]^+, -62[NO3]^-, -26[CN]^-, -35[Cl]^-, -96[SO4]^-, -46[NO2]^-, -97[HSO4]^- as the potential reference ions for ambient aerosols.

Step 2: a set of reference ions was chosen from the potential reference ion pool for each spectrum. The selection was based on the absolute ion intensity of the reference ions in this spectrum. They must be greater than a threshold, e.g. we set 15 a.u. for ambient aerosol and 8 a.u. for sea spray aerosol, respectively. A particle was discarded from the spectra database if it did not have enough reference ions (the minimum number of reference ions was set to be 5) in either positive or negative mass spectrum.

Step 3: the reference ions were used to calibrate m/z for mass spectra of each particle. As introduced before, a HR-SPAMS spectrum consists of a number of bins. The measured m/z bin values of the reference ions mentioned in the Step0 were calibrated based on their theoretic (or true) m/z bin values. A linear regression between the two set of variables (measured vs. theoretic m/z bin values) was conducted, and two calibration parameters (a slope and an intersect) can be obtained. Then we used these parameters to make the calibration for every bin value in this mass spectra. Finally, the m/z of the whole spectrum had been corrected. Thus, we assigned a m/z bin value to each corrected m/z based on proximity principle. Finally, mass spectra with well calibrated bin value can be obtained for each single particle.

A GUI program for this automatic linear calibration method had been developed for the sake of easy use (Fig. 1). The MATLAB codes for this GUI and the automatic linear calibration method are open access and available at https://github.com/zhuxiaqtian-fdu/zhuxiaqtian-fdu.

3.2 Evaluation of the calibration method

In this study, a total of 5,130 sea spray aerosol particles and 5,007 ambient aerosol particles
were analyzed. And 4,624 sea spray particles and 1,409 ambient particles were successfully calibrated. As some fraction of particles had been filtered because their mass spectrum did not have 5 or more reference peaks to conduct the calibrations. To mitigate this problem, we proposed some adjustments in the next section. Figure 2 shows that the calibration curves for a random selected sea spray aerosol particle and ambient aerosol particle. The adj-R² coefficients of both calibration curves are equal to ~1, demonstrating that this calibration method is effective and accurate. All the slopes and intercepts of the linear calibration can be found in the Fig.S3 and Fig.S4.

3.3 Automatic linear calibration method with a larger reference ion pool

It is important to note that a large number of ambient particles were filtered because their spectra did not have 5 or more reference peaks to conduct calibrations. Especially, only ~29.0% of total ambient particles had sufficient number of reference ions in their positive spectra. To solve this problem, extra reference ions, including 67[VO]⁺, 67[C₅H₇]⁺, 89[C₇H₅]⁺, 89[Na₂BO₂]⁺, 102[C₈H₆]⁺, 102[CaNO₃]⁺, were added into the original positive reference ion pool. Obviously, these ions share the same integer m/z value with other ions. We needed to identify them using additional information other than their integer m/z values.

The specific reference ions was determined by their coarsely-calibrated m/z. Table S2 shows that the m/z deviation ranges of the reference positive ambient ions in coarsely-calibrated spectra before automatic linear calibration were around 0.011~0.048, while the m/z differences between 67[VO]⁺ and 67[C₅H₇]⁺, 89[C₇H₅]⁺ and 89[Na₂BO₂]⁺, 102[C₈H₆]⁺ and 102[CaNO₃]⁺ were 0.1213, 0.0622, and 0.083, respectively, which were larger than the m/z deviations of these reference ions in coarsely-calibrated spectra. Therefore, the coarsely-calibrated spectra can be used to determine these specific reference ions. With these additional potential reference ions, a total of 2490 ambient particles were calibrated,
4. Application to atmospheric aerosols measurement

4.1 HR-SPAMS measurement of sea spray aerosol

SPAMS data usually contains a large number of individual mass spectra. It is impossible to manually analyze every spectrum from a large dataset. Averaging a number of mass spectra is often preferred. However, to obtain averaged high-resolution spectrum, each spectrum must be well calibrated. Therefore, it would be very interesting to see what new information can be obtained from HR-SPAMS measurement of aerosols with the automatic calibration method. Figure 4 reports the average positive and negative mass spectra for the laboratory generated sea spray aerosols and the error limits mean the concluded accepted error range. Similar to the low-resolution sea spray aerosol mass spectra, they contain major peaks of Na\(^+\), Mg\(^+\), K\(^+\), Na\(_2\)Cl\(^+\), CN\(^-\), Cl\(^-\), CNO\(^-\), NaCl\(^-\), NaCl\(_2\)\(^-\) and MgCl\(_3\)\(^-\), as well as many smaller peaks, such as Ca\(^+\), SiO\(_2\)/SiO\(_3\)\(^-\), and KCl\(_2\). With the improved m/z measurement, many peaks, which cannot be determined by integer resolution mass spectra, now can be clearly identified (Table 2). For example, the ion with m/z at 27.0267 is C\(_2\)H\(_3\)\(^+\) rather than Al\(^+\). The ion with m/z at 76.9336 is CaCl\(^+\) rather than C\(_6\)H\(_5\)\(^+\). And some sulfur containing organic ions, such as CS\(^+\), can be determined. Surprisingly, we can identify the presence of HCO\(_2\)\(^-\) and CaCO\(_3\)\(^-\), demonstrating that carbon hydrates are contained in sea spray aerosols.

4.2 HR-SPAMS measurement of atmospheric aerosol

Laboratory generated sea spray aerosol can be viewed as a relatively simple aerosol system, while the chemical compositions of ambient aerosols are much more complicated. Figure 5 shows the averaged HR-SPAMS mass spectra of the ambient aerosols sampled at Fudan University Jiangwan Campus on May 29\(^{th}\), 2019. With the improved m/z measurement, many organic ions, such as C\(_x\), C\(_x\)H\(_y\), and C\(_x\)H\(_y\)O\(_z\) can be directly identified (Table 3). Also, we can separate the organic and inorganic species more directly with the high mass resolution. For instance, C\(_6\)H\(_8\)\(^+\) can be clearly distinguished from possible interference of...
Ca\(^{2+}\), TiO\(^{2+}\) and NaKO\(^{+}\). C\(_{10}\)H\(_{-}\) can also be identified from possible assignment of NaSO\(_{4}^{-}\) etc. More importantly, 139[C\(_{2}\)H\(_{3}\)O\(_{5}\)S\(^{-}\)] (the theoretical m/z value: -138.97) can be clearly distinguished from other possible assignments, such as 139[C\(_{1}\)H\(_{7}\)] with the m/z value of -139.55 and 139[AsO\(_{4}^{-}\)] with the m/z value of -138.90. Moreover, 153[C\(_{3}\)H\(_{5}\)O\(_{5}\)S\(^{-}\)] with theoretical m/z value of -152.986 can be distinguished from other possible assignments, such as 153[C\(_{12}\)H\(_{9}\)] with m/z value of -153.070 and 153[Na\(_{2}\)Cl\(_{3}\)] with m/z value of -152.883. These two important organic ion peaks have been suggested to be the characteristic ion peaks for the organosulfates in secondary organic aerosols (Surratt et al., 2010; Surratt et al., 2007).

4.3 Time variation of HR-SPAMS measurement

With the high mass resolution of HR-SPAMS and enhanced m/z calibration, we were able to obtain an average mass spectrum from many particles. The accurate m/z values in the average mass spectrum can be used to separate peaks with close m/z and track their intensity variations. Here we conducted a time variation measurement for ambient aerosol from 11:00 on May 29\(^{th}\) to 11:00 on May 30\(^{th}\)2019. We selected first 500 particles collected by SPAMS during every hour for elemental analysis. Figure 6 shows that the peak at m/z 41 has a bimodal structure, whose m/z were at 40.9546±0.0105 and 41.0194±0.0105, respectively. Thus the peak with the smaller m/z is an isotope of K\(^{+}\)(theoretical m/z value of \(^{41}\text{K}^{+} = 40.96182\), theoretical m/z bin value of \(^{41}\text{K}^{+} = 40.9667\); this peak also follows the isotopic pattern of K) and the other peak should be C\(_{3}\)H\(_{5}\)\(^{+}\)(theoretical m/z value 41.03913, theoretical m/z bin value 41.03). Figure 6 shows that HR-SPAMS was able to separately measure the time series of these two peaks with a small m/z difference. In contrast, it is impossible for a low resolution (LR)-SPAMS to provide such detailed time variation measurement of these peaks.

4.4 Particle classification by ART-2a

Adaptive resonance theory neural network (ART-2a) is a widely-used method to classify particles based on the similarity among their mass spectra (Song et al., 1999). Here we make a comparison of ART-2a classification between the HR-SPAMS data and traditional
low resolution (LR)-SPAMS data. Particles with the positive and negative spectra were analyzed by ART-2a with a learning rate of 0.05, a vigilance factor of 0.7, and an iteration number of 20. The previous ambient aerosol SPAMS dataset (1,400 particles) was used for the matrix size of the ART-2a is around $7 \times 10^7$. The LR-SPAMS data, whose m/z was at integer level, was generated by summing high resolution SPAMS peaks in each integer m/z bin. The classification results show that the HR-SPAMS data was grouped to 93 categories and the top 45 categories accounted for 96 percent of all particles. The particle number of the first eight categories was 122, 101, 99, 86, 82, 70, 68 and 60 respectively. In contrast, the LR-SPAMS data was only grouped to 33 categories in total and the top 20 categories accounted for the 96 percent of all particles. The particle number of the first eight categories was 170, 118, 107, 107, 106, 92, 90 and 88 respectively. The detailed results can be found in the Fig.S5-S6. Obviously, ART-2a classification of high resolution SPAMS data generated more particle categories. This is mainly because HR-SPAMS mass spectra can differentiate peaks with close m/z, which may be viewed as one peak in LR-SPAMS data. The ART-2a classification of the HR-SPAMS results (Fig.S6) show that the signal at 23$[\text{Na}^+]$ in Type 2HR was stronger than Type 1HR while the signals at 26$[\text{CN}^-]$ and 42$[\text{CNO}^-]$ were weaker in Type1 HR. Meanwhile the averaged mass spectra of the Type 2HR showed the presence of 206$[\text{Pb}^+]$, 207$[\text{Pb}^+]$ and 208$[\text{Pb}^+]$, which are known to be harmful to human health (Das et al., 2018; Peng et al., 2020). Furthermore, particles of Type 2HR containing abundant secondary inorganic components like $[\text{NO}_2^-]$, $[\text{NO}_3^-]$ and $[\text{SO}_4^+]$, which originated from the aerosol aging processes (Dall'Osto and Harrison, 2012; Ma et al., 2016). In contrast, these two first particle types were lumped together into Type 1LR in the LR-SPAMS classification results (Fig.S5). Due to the merge of these two particle types, $[\text{Pb}^+]$ ions were not significant. Meanwhile, Type 3,4,5HR classification results contain strong signals at 26$[\text{CN}^-]$, 42$[\text{CNO}^-]$, 46$[\text{NO}_2^-]$, 62$[\text{NO}_3^-]$ and 97$[\text{HSO}_4^-]$, suggesting that these three types were from biomass burning or residential cooking burning. $[\text{K}^+]$ is also another feature of this type particle emission (Bi et al., 2011; Hudson et al., 2004). There were obvious relative ion intensity differences at 26$[\text{CN}^-]$, 42$[\text{CNO}^-]$, 46$[\text{NO}_2^-]$ and 62$[\text{NO}_3^-]$ among these types, which implied that these three particle types
might be from different burning sources or experienced different levels of aging (Luo et al., 2020). While these three particle types were lumped together as Type 2LR. This critical information which could be potentially used to distinguish particle sources and aging processes was lost. Additionally, Type 7HR can be assigned as ECOC type, based on its strong signals at \([C_\text{x}^+], [C_\text{x}H_y^+]\) and \([C_\text{x}H_yO_z^+]\). Particles of this type may come from the primary emission sources, and the emitted black carbon particles would also form this type particles after absorbing some low volatile organic compounds in the atmosphere (Sodeman et al., 2005). 97[HSO\(_4\)]\(^-\) can be observed to have a weaker signal than 62[NO\(_3\)]\(^-\) and 46[NO\(_2\)]\(^-\), which implied that the secondary reaction of SO\(_2\) as the precursor of HSO\(_4\)\(^-\) was not significant in the particle surface for ECOC type particles in this study (SULLIVAN and PRATHER, 2007). In contrast, the classification results of the LR-SPAMS were not so clear and generated less particle types. Given HR-SPAMS spectra have much more detailed chemical information about particles, we would propose that the ART-2a classification of HR-SPAMS might be more accurate.

5. Conclusion

An automatic linear calibration method had been developed for data analysis of high-resolution SPAMS data. This technique can significantly improve the m/z accuracy of SPAMS spectra for atmospheric aerosol samples. The analysis of HR-SPAMS data for laboratory generated sea spray aerosols shows many details of its chemical compositions. For example, many organic ions, such as \(C_2H_3^+\) and \(CS^+\), can be directly determined. The chemical compositions of ambient aerosols are much more complicated. It is found that, besides major ions (e.g. Na\(^+\), K\(^+\), Ca\(^+\), Fe\(^+\), Cl\(^-\), CN\(^-\), NO\(_3^-\) and HSO\(_4^-\)), \(C_\text{x}H_y\), \(C_\text{x}H_yO_z\) and CNO\(^-\) can be identified. With this method, HR-SPAMS can also determine the time series of organic and inorganic peaks, whose m/z are very close to each other (e.g. 41K\(^+\) with the theoretical m/z value at 40.96182 and \(C_3H_5^+\) with the theoretical m/z value at 41.03913). Important organic ion peaks, such as tracer peaks for secondary organic matter like 139[\(C_2H_3O_5S^-\)] and 153[\(C_3H_5O_5S^-\)], can be identified. More importantly, our ART-2a classification from HR-SPAMS dataset clearly showed a particle type containing heavy metals like Pb\(^+\), which was obviously ignored in the ART-2a classification from the LR-
SPAMS dataset. More particle types were generated by the ART-2a classification of HR-SPAMS data compared to that of LR-SPAMS data, as the original biomass burning particle type can be divided into three more detailed types based on the different signals of 26[CN\(^-\)], 42[CNO\(^-\)], 46[NO\(_2^+\)] and other organic species like C\(_x\)H\(_y\) and C\(_x\)H\(_y\)O\(_z\), implying different aerosol aging processes or burning conditions. Such detailed information may be critical to study the aging processes and source appointment of atmospheric aerosols. There is a deficiency of this HR-SPAMS calibration method, which has been showed that some fraction of particles cannot be calibrated due to the presence of weak signals of the marker ions. It can be mitigated by applying some additional marker ions. All the automatic linear calibration method codes specific for HR-SPAMS are open access and can be found at https://github.com/zhuxiaoqiang-fdu/zhuxiaoqiang-fdu. And we proposed the principle of this calibration method can be adopted in other aerosol mass spectrometers.

Author Contribution
Y. X and X. W. supervised this study. X. W. and S. Z. designed the calibration and data analysis methods. S. Wang, and S. Z. performance the sea spray aerosol and ambient aerosol experiment. S. Z. wrote the open source code for calibration and data analysis of the single particle mass spectra and made the GUI program with suggestions from X. W. and X. Y.. X. W. and S. Z. prepared the manuscript with contributions from all co-authors.

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Competing interests
The authors declare that they have no conflict of interest.
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Figure Captions

Figure 1. The GUI program for HR-SPAMS calibration

Figure 2. Linear calibration with reference ion peaks

Figure 3. Probability distributions of the marker peak locations before and after Automatic Linear Calibration (AL-Cal) for (a) sea spray aerosol and (b) ambient aerosol

Figure 4. Averaged positive and negative mass spectra of sea spray aerosols

Figure 5. Averaged positive and negative mass spectra of ambient aerosols

Figure 6. Time series of peak intensities at m/z 40.95 and m/z 41.01
Table 1. Possible peak assignments for the m/z of reference ions for sea spray aerosol and ambient aerosol

| Unit mass resolution m/z | Possible species (Sea spray aerosol) | Unit mass resolution m/z | Possible species (Ambient aerosol) |
|--------------------------|--------------------------------------|--------------------------|-------------------------------------|
| +24                      | Mg$^+$ C$_2^+$                       | +39                      | K$^+$ C$_3$H$_3^+$                  |
| +39                      | K$^+$ C$_3$H$_3^+$                   | +56                      | Fe$^+$ Si$_2^+$ CaO$^+$ KOH$^+$     |
| +81                      | Na$_2$Cl$^+$ Br$^+$ C$_6$H$_9^+$     | -26                      | CN$^-$ BO$^-$ C$_2$H$_2^-$          |
| +113                     | K$_2$Cl$^+$ C$_9$H$_5^+$             | -62                      | NO$_3^-$ C$_5$H$_2^-$               |
| +115                     | K$_2$Cl$^+$ C$_9$H$_7^+$             | -96                      | SO$_4^-$ BrOH$^-$                   |
| -26                      | CN$^-$ BO$^-$ C$_2$H$_2^-$           | -97                      | HSO$_4^-$ C$_8$H$_7^-$ BrO$^-$ NaCl$_2^-$ |
|                           | Cl$^-$ C$_3$H$^-$                    |                          | H$_2$PO$_4^-$                       |
| -37                      | BO$_2^-$ CNO$^-$                     |                          |                                     |
| -42                      | MgCl$_3^-$ C$_{10}$H$_9^-$ (C$_3$H$_7$)$_2$C$_2$H$_5^-$ |
|                           | CaCl$_2$OH$^-$                       |                          |                                     |
| -129                     |                                      |                          |                                     |
| -131                     | MgCl$_3^-$                           |                          |                                     |
Table 2. Peak identification of important chemical species in sea spray aerosols. The first column is the measured m/z for peaks. The second and third columns show the theoretical m/z bin value and theoretical m/z value of most possible species for each peak.

| Measurement m/z (positive) | Possible species | Possible species | Measurement m/z (negative) | Possible species | Possible species |
|----------------------------|------------------|------------------|----------------------------|-----------------|-----------------|
| 22.993                     | Na⁺(22.993)      | Na⁺(22.98977)    | 15.0344                    | CH₃⁺(15.0216)   | CH₃⁺(15.02348)  |
| 23.9829                    | Mg⁺(23.9829)     | Mg⁺(23.98505)    | 34.9641                    | Cl⁻(34.9641)    | Cl⁻(34.96885)   |
| 27.0267                    | C₂H₃⁺(27.0267)   | C₂H₃⁺(27.02348)  | 41.9864                    | CNO⁻(41.9971)   | CNO⁻(41.99799)  |
| 38.9672                    | K⁺(38.9672)      | K⁺(38.96371)     | 25.0163                    | C₂H(25.0081)    | C₂H(25.00783)   |
| 39.9711                    | Ca⁺(39.9607)     | Ca⁺(39.96259)    | 38.0024                    | C₃H₂(38.0126)   | C₃H₂(38.01565)  |
| 43.9614                    | CS⁺(43.9723)     | CS⁺(43.9721)     | 44.9883                    | HCO₂⁻(44.9994)  | HCO₂⁻(44.99767) |
| 45.983                     | Na₂⁺(45.983)     | Na₂⁺(45.97954)   | 49.002                     | C₄H(49.0135)    | C₄H(49.00783)   |
| 59.9569                    | SiO₂⁻(59.9696)   | SiO₂⁻(59.96677)  | 57.9574                    | NaCl⁻(57.9574)  | NaCl⁻(57.95865) |
| 71.9872                    | C₆⁺(72.0012)     | C₆⁺(72)          |                            |                 |                 |
| 80.9438                    | Na₂Cl⁺(80.9438)  | Na₂Cl⁺(80.94839) | 63.9574                    | SO₂⁻(63.9574)   | SO₂⁻(63.96191)  |
| 112.898                    | K₂Cl⁺(112.898)   | K₂Cl⁺(112.89627) | 75.9498                    | SiO₃⁻(75.9642)  | SiO₃⁻(75.96196) |
| 138.89                     | Na₃Cl₂⁺(138.907) | Na₃Cl₂⁺(138.90702) | 79.9547                    | SO₃⁻(79.9547)   | SO₃⁻(79.95683)  |
| 140.897                    | Na₃Cl₂⁺(140.897) | Na₃Cl₂⁺(140.90407) | 80.9015                    | Br⁺(80.9164)    | Br⁺(80.91629)   |
|                            |                  |                  | 85.9484                    | NaPO₂⁻(85.9484) | NaPO₂⁻(85.95337) |
|                            |                  |                  | 99.9499                    | CaCO₃⁻(99.9499) | CaCO₃⁻(99.94735) |
|                            |                  |                  | 109.917                    | CaCl₂⁻(109.9)   | CaCl₂⁻(109.9003) |
|                            |                  |                  | 128.901                    | MgCl₁⁻(128.883) | MgCl₁⁻(128.89161) |
Table 3. Peak identification of important chemical species in ambient aerosols. The first column is the measured m/z for each peak. The second and third columns shows the theoretical m/z bin value and theoretical m/z value of most possible species for each peak.

| Measurement m/z(positive) | Possible species(theoretical m/z bin value) | Possible species(theoretical m/z value) | Measurement m/z(negative) | Possible species(theoretical m/z bin value) | Possible species(theoretical m/z value) |
|--------------------------|-----------------------------------------------|---------------------------------------------|--------------------------|-----------------------------------------------|---------------------------------------------|
| 22.993                   | Na⁺(22.993)                                    | Na⁺(22.98977)                               | 15.0408                  | CH₃⁻(15.0216)                                 | CH₃⁻(15.02348)                             |
| 23.9991                  | C₂⁺(23.9991)                                   | C₂⁺(24)                                    | 16.0091                  | O⁻(15.9959)                                   | O⁻(15.9942)                                |
| 25.002                   | C₂H⁺(25.0103)                                  | C₂H⁺(25.00783)                             | 17.0145                  | OH⁻(17.0009)                                  | OH⁻(17.00275)                              |
| 26.0087                  | C₂H₂⁺(26.0171)                                 | C₂H₂⁺(26.01565)                           | 26.0078                  | CN⁻(25.9994)                                  | CN⁻(26.00307)                              |
| 30.0171                  | NO⁺(29.999)                                    | NO⁺(29.99799)                              | 31.987                   | O₂⁻(31.987)                                   | O₂⁻(31.9894)                               |
| 35.9925                  | C₃⁺(36.0023)                                   | C₃⁺(36)                                    | 34.9641                  | Cl⁻(34.9641)                                  | Cl⁻(34.96885)                              |
| 36.9976                  | C₃H⁺(37.0076)                                  | C₃H⁺(37.00783)                             | 41.9971                  | CNO⁻(41.9971)                                 | CNO⁻(41.99799)                             |
| 38.0065                  | C₃H₂⁺(38.0166)                                 | C₃H₂⁺(38.01565)                           | 45.9897                  | NO₂⁻(45.9897)                                 | NO₂⁻(45.99291)                             |
| 38.9672                  | K⁺(38.9672)                                    | K⁺(38.96371)                               | 47.9911                  | C₄⁺(47.993)                                   | C₄⁺(48)                                    |
| 47.993                   | C₄⁺(47.993)                                    | C₄⁺(48)                                    | 61.9808                  | NO₃⁻(61.9938)                                 | NO₃⁻(61.98783)                             |
| 48.9911                  | C₄H⁺(49.0026)                                  | C₄H⁺(49.00783)                             | 71.0014                  | C₃H₅O₂⁻(71.0153)                              | C₃H₅O₂⁻(71.01332)                         |
| 49.9994                  | C₄H₂⁺(50.0111)                                 | C₄H₂⁺(50.01565)                           | 78.9548                  | PO₃⁻(78.9548)                                 | PO₃⁻(78.95852)                             |
| 55.9443                  | Fe⁺(55.932)                                    | Fe⁺(55.93494)                              | 79.94                   | SO₃⁻(79.9547)                                 | SO₃⁻(79.95683)                             |
| 59.9951                  | C₅⁺(59.9951)                                   | C₅⁺(60)                                    | 80.946                   | HSO₃⁻(80.9609)                                | HSO₃⁻(80.96466)                           |
| 60.9946                  | C₅H⁺(61.0074)                                  | C₅H⁺(61.00783)                            | 95.9825                  | SO₄⁻(95.9502)                                 | SO₄⁻(95.95175)                             |
| 62.0023                  | C₅H₂⁺(62.0152)                                 | C₅H₂⁺(62.01565)                            | 96.9546                  | HSO₄⁻(96.9546)                                | HSO₄⁻(96.95958)                            |
| 72.0012                  | C₆⁺(72.0012)                                   | C₆⁺(72)                                    | 121.01                   | C₁₀H⁺(121.01)                                 | C₁₀H⁺(121.00783)                           |
| 84.0108                  | C₇⁺(83.9957)                                   | C₇⁺(84)                                    | 122.01                   | C₁₀H₂⁺(122.01)                                | C₁₀H₂⁺(122.01565)                         |
| 207.976                  | Pb⁺(207.967)                                   | Pb⁺(207.97664)                             | 134.008                  | C₁₁H₂⁺(134.008)                               | C₁₁H₂⁺(134.01565)                         |
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