Methodology Using a Portable X-Ray Fluorescence Device for On-Site and Rapid Evaluation of Heavy-Atom Contamination in Wounds: A Model Study for Application to Plutonium Contamination

Hiroshi Yoshii1*, Kouta Yanagihara1,2, Hitoshi Imaseki1, Tsuyoshi Hamano1, Hirokuni Yamanishi3, Masayo Inagaki3, Yasuhiro Sakai2, Nobuyuki Sugiura4, Osamu Kurihara1, Kazuo Sakai1

1 Research Center for Radiation Emergency Medicine, National Institute of Radiological Science, Chiba, Chiba, Japan, 2 Department of Physics, Faculty of Science, Toho University, Funabashi, Chiba, Japan, 3 Atomic Energy Research Institute, Kinki University, Higashi-Osaka, Osaka, Japan, 4 Radiation Environmental Effects Research Center, Nuclear Safety Research Association, Shinbashı, Minato, Tokyo, Japan

Abstract

Workers decommissioning the Fukushima-Daiichi nuclear power plant damaged from the Great East Japan Earthquake and resulting tsunami are at risk of injury with possible contamination from radioactive heavy atoms including actinides, such as plutonium. We propose a new methodology for on-site and rapid evaluation of heavy-atom contamination in wounds using a portable X-ray fluorescence (XRF) device. In the present study, stable lead was used as the model contaminant substitute for radioactive heavy atoms. First, the wound model was developed by placing a liquid blood phantom on an epoxy resin wound phantom contaminated with lead. Next, the correlation between the concentration of contaminant and the XRF peak intensity was formulated considering the thickness of blood exiting the wound. Methods to determine the minimum detection limit (MDL) of contaminants at any maximal equivalent dose to the wound by XRF measurement were also established. For example, in this system, at a maximal equivalent dose of 16.5 mSv to the wound and blood thickness of 0.5 mm, the MDL value for lead was 1.2 ppm (3.1 nmol). The radioactivity of 239Pu corresponding to 3.1 nmol is 1.7 kBq, which is lower than the radioactivity of 239Pu contaminating puncture wounds in previous severe accidents. In conclusion, the established methodology could be beneficial for future development of a method to evaluate plutonium contamination in wounds. Highlights: Methodology for evaluation of heavy-atom contamination in a wound was established. A portable X-ray fluorescence device enables on-site, rapid and direct evaluation. This method is expected to be used for evaluation of plutonium contamination in wounds.

Introduction

The Fukushima-Daiichi nuclear power plant damaged from the Great East Japan Earthquake and resulting tsunami is currently being decommissioned. In the daunting task of decommissioning the plant, workers are at risk of injury with possible contamination from toxic radioactive heavy atoms including plutonium. Nuclear power plants or α-emitter handling facilities are equipped with survey meters for α-particle emissions [1]. Plutonium in the wound is, however, hard to detect because plutonium rarely emits γ-rays, and α-particles emitted from plutonium cannot pass through blood exiting the wound [2]. Genicot et al. [3] have developed a method for estimating the quantity of 239Pu from the intensity of weak 59.5 keV γ-rays emitted spontaneously from co-localized 241Am using a germanium semiconductor detector with calculations based on the known abundance ratio of 239Pu and 241Am. However, this passive detection method takes approximately 15 min. For more rapid diagnosis and initiation of the treatment of the wound, new methods for on-site and rapid evaluation of plutonium contamination in a wound are required.

To meet this requirement, we focused on X-ray fluorescence (XRF) analysis, which can be applied to elemental analysis. The intensity of the XRF peak depends on the number of corresponding atoms, and the number of atoms per unit radioactivity (i.e., per 1 Bq) is directly proportional to the half-life of the radionuclide. Thus, XRF analysis for a long-half-life radionuclide, such as plutonium, would be effective. Because XRF analysis is an active detection technique, the measurement time for XRF analysis is generally shorter than the passive detection of photons, X- and γ-rays, emitted spontaneously. Additionally, portable XRF devices can provide on-site measurement. Thus, XRF analysis using portable devices can potentially be used for on-site and rapid detection of plutonium-contaminated wounds.

XRF spectrometry has been used in the detection of contaminants in human tissue [4–9] and their phantoms [10–14]; however, it has not yet been applied to wounds. The first step in
the application to wounds is to develop an XRF analysis methodology that considers the attenuation of X-rays by blood. The purpose of the present study is, therefore, to develop the methodology for on-site and rapid evaluation of contaminants in the wound using a portable XRF device considering the blood thickness. This study presents a methodology for evaluation of heavy-atom contamination in wounds using stable lead as the model contaminant, because lead is a familiar heavy atom and the XRF peak energies for Pb Lα (10.55 keV) and Pb Lβ (12.61 keV) are close to those for Pu Lα (14.20 keV) and Pu Lβ (18.20 keV), respectively [15]. In the absence of a standard heavy-wound model, we developed models for dry and bleeding wounds in this study. The contaminated wound phantom comprises an epoxy resin contaminated with stable lead, since the resin, with a density that is similar to that of human soft tissues, has been previously used to imitate the human skin [14]. The developed wound model has blood phantoms containing liquid blood combined with epoxy resin wound phantoms containing stable lead, for the XRF analysis, and developed the methodology to evaluate heavy-atom contamination in a wound considering the thickness of blood exiting the wound.

Materials and Methods

Bleeding wound model

The bleeding wound model with stable-lead contamination was constructed by placing the blood phantom on the epoxy resin wound phantom described below (Fig. 1b). The epoxy resin wound phantom without the blood phantom can be used as a dry wound model (Fig. 1a).

Epoxy resin wound phantoms were prepared by mixing epoxy resin (crystal resin, Nissin Resin Co., Ltd., Yokohama, Japan) with different concentrations of lead in white oil paint. In this study, white oil paint (Silver White, Kusakabe Co., Ltd., Saitama, Japan) containing 60.5% lead (from manufacturer’s material safety data sheets) was used as the standard material. Solutions with different concentrations of lead (0, 200, 500, 1000, 1500 and 2000 ppm = μg/g) were prepared by precisely mixing appropriate volumes of the standard material and the solvent for oil paints (Odorless Petroleum, Holbein Works, Ltd., Osaka, Japan). Resulting solutions were mixed extensively with epoxy resin (paint solution-to-epoxy resin ratio = 1/99) to provide final lead concentrations of 0, 2, 5, 10, 15 and 20 ppm. Both the diameter and height of the molded epoxy resin wound phantoms were 10 mm (Fig. 1a). The density of the epoxy resin wound phantom was 1.06 g cm⁻³, which is similar to that of human soft tissue.

To investigate the effect of thickness of blood layer on the peak intensity of XRF, phantoms of blood with different thicknesses were constructed. For this, a circular piece (10-mm diameter) was cut out in each of the several acrylic plates (thickness = 0.5, 1.0, 1.5 and 2.0 mm). This 10-mm diameter opening could accommodate the irradiation field of the XRF device (8-mm diameter). One side of this circular aperture was sealed with a 7.5-μm-thick kapton film. Lead-free blood from mice (Japan SLC, Inc., Shizuoka, Japan) was then loaded into this opening. Finally, the opening on the other side was sealed by a second kapton film (Fig. 1b). Therefore, the constructed blood phantom had both sides of the aperture sealed by kapton films with a layer of blood enclosed in it. The specific gravity, which is a main factor affecting X-ray attenuation of mouse is similar to that of human blood [16,17]. Furthermore, composition of elements identified by XRF in the region of this study, such as Fe, Zn, and Cu, are similar for both mouse and human blood [18–21]. Therefore, the attenuation of X-rays by mouse blood will be similar to that by human blood. In this study, lead was added to the epoxy resin, not to the blood.

This was done to reflect the actual situation in radiation emergency. Typically, after an injury, the blood is washed off during the preliminary cleaning of the wound. Evaluation of contaminants adhering to a wound is performed subsequently, sometimes even while the fresh blood oozes out. To represent this situation, we constructed the bleeding wound model by placing the heavy metal-free blood phantom on the stable lead containing epoxy resin wound phantom. Thus, bleeding and dry wound models imitating wound with or without newly oozing blood were constructed.

XRF apparatus

A portable XRF device, XL3t-950S (Thermo Fisher Scientific Inc.), which is easy to handle, was used in the present study. The silicon drift detector was installed in the XL3t-950S instead of the Si PIN diode used in the XL3 [8], and a silver (Ag) anode was employed. The diameter of the measurement range of the device is approximately 8 mm. In our measurements, the X-ray tube voltage was set to 50 kV, the current was set automatically, and the main filter of the XL3t-950S was selected. According to the XL3t-950S user’s manual, an equivalent dose to skin is less than 16.5 mSv for a 5 s exposure time (accumulation time) when the device is applied to the skin surface. In the present study, the maximal equivalent dose to the wound in a human was estimated from the accumulation time in the measurement using the bleeding wound model.

Procedure

Accumulation times were set to 5, 10, 15 and 20 s. Because exposure dose to the wound is directly proportional to exposure time, the maximal equivalent doses to the wound are estimated as 16.5, 33.0, 49.5 and 66.0 mSv for the accumulation times of 5, 10, 15 and 20 s, respectively. XRF analysis was performed for 120
cases, the combination of six levels of lead concentration \((c)\), five kinds of the blood phantom thickness \((b)\) and four levels of maximal equivalent doses to the wound \((d)\) estimated based on the accumulation time \((n = 4, \text{ for each case})\). The measurement cases in the present study are indicated by these parameters; for example, \((c = 2, t = 0.5, d = 16.5)\) indicates the measurement using the bleeding wound model constructed by the epoxy resin wound phantom containing 2-ppm stable lead \((c = 2)\) and the blood phantom, whose thickness is 0.5 mm \((t = 0.5)\), at a maximal equivalent dose to the wound estimated as 16.5 mSv \((d = 16.5)\) obtained from the accumulation time of 5 s. Here, \(t = 0.0\) mm means that no blood phantom was placed on the epoxy resin wound phantom at each \(c\) and \(d\). As described above (Fig. 1a), this is the dry wound model. In the present study, the gross intensity of the measured XRF peak was obtained as the integrated intensity rather than the result of Gaussian fitting, since Pb L\(_a\) and L\(_\beta\) peaks have asymmetric shapes due to the presence of unresolved subcomponent peaks (such as L\(_{\alpha 1}\) and L\(_{\alpha 2}\) for L\(_a\) peak and L\(_{\beta 1}\), L\(_{\beta 3}\), L\(_{\beta 4}\) and L\(_{\beta 5}\) for L\(_\beta\) peak) [15,22]. Background spectra are the spectra for the model using the epoxy resin wound phantom without lead, i.e., the spectra for the cases of \(c = 0\) ppm at each \(t\) and \(d\). Because there are five different blood thicknesses and four different maximal exposure doses to the wound, the spectra for 20 cases can be regarded as the background spectra, respectively. Here, the background intensity for a specific peak (e.g., Pb L\(_a\) peak) is defined as the gross intensity in the background spectrum at the energy range where the corresponding peak (Pb L\(_a\) peak) was found in the spectra for the model using the lead-contaminated epoxy resin wound phantom. The net intensity of the peak was calculated after subtracting the background intensity from gross intensity.

**Results**

The measured spectra

A representative XRF spectrum is shown in Fig. 2a for the case of \((c = 20, t = 0, d = 66.0)\). Device-derived peaks, including Ni K\(_a\), Ni K\(_\beta\), Mo K\(_\beta\) and its Compton scattering, Mo K\(_\beta\) and its Compton scattering, Ag K\(_a\) and Ag K\(_\beta\), were found in the measured spectrum. Since the anode of X-ray tube is silver, the intense peaks attributed to silver were clearly observed. The intensities of device-derived peaks are reduced by putting the blood phantom on the epoxy resin wound phantom (Fig. 2b). The intensities of device-derived peaks for the case of \((c = 20, t = 2.0, d = 66.0)\) are similar to those for the case of \((c = 0, t = 0, d = 66.0)\). This indicates that the intensities for the device-derived peaks were not dependent on the blood thickness. Similarly, the background intensities at the Pb L\(_a\) peak for each blood thickness were the same within the margin of error (spectra not shown).

**Correlations between intensity of the XRF peak and lead concentration, dose to wound, and blood thickness**

The gross intensities of the Pb L\(_a\) peak in the spectra for the dry wound model were plotted against lead concentration in the epoxy resin wound phantom for each maximal equivalent dose to the wound (Fig. 2c). In Fig. 2c, the background intensities at the Pb L\(_a\) peak for each maximal exposure dose to the wound are given as the intercept. Fig. 2c shows that the background intensities at the Pb L\(_a\) peak are directly proportional to the maximal equivalent dose to the wound \((d)\). Since both the maximal equivalent dose to the wound and background intensity are directly proportional to the measurement time independently, it was reasonable to assume a direct relationship between background intensity and the maximal equivalent dose to the wound. The net intensity of the Pb L\(_a\) peak can be obtained by subtraction of the background intensity shown as the intercept in Fig. 2c from the gross intensity of the Pb L\(_a\) peak. The net intensity of the Pb L\(_a\) peak was directly proportional to both the lead concentration in the epoxy resin wound phantom \((c)\) and the maximal equivalent dose to the wound \((d)\). Generally, the net intensity of the XRF peak is directly proportional to the concentration of the target element and the accumulation time [22]. Since the maximal equivalent dose to the wound is directly proportional to the accumulation time, it is gratifying that the net intensity is directly proportional to \(c\) and \(d\).

The curves of the gross intensity of the Pb L\(_a\) peak versus lead concentration in the epoxy resin wound phantom for each blood thickness at \(d = 66.0\) mSv are drawn in Fig. 2d. The gross intensity of the Pb L\(_a\) peak was inversely correlated to the blood thickness for each concentration of lead in the epoxy resin wound phantom. There are no differences among the background intensities of the Pb L\(_a\) peak given as the intercept for several blood thicknesses.

**Discussion**

In the past three decades, XRF has been used in the analysis of human tissue and their phantoms [4–14]. For example, Nie et al. developed a method to measure lead contamination in bone using a portable XRF spectrometer for the evaluation of human chronic exposure to lead [8]. However, XRF has not yet been used in the analysis of wounds. In this study, we described a portable XRF device-based method for on-site and rapid evaluation of heavy-atom contamination in a wound.

As shown in Fig. 2c, the background intensity at the Pb L\(_a\) peak is directly proportional to \(d\) and can be represented by \(I_{od}\), where the proportionality constant, \(I_b\), is the background intensity when \(d = 1\) mSv. Similarly, the net intensity of the Pb L\(_a\) peak is directly proportional to both \(d\) and \(c\) and is represented by \(I_{dc}\), where the proportionality constant, \(I_c\), is the net intensity of the Pb L\(_a\) peak when \(d = 1\) mSv and \(c = 1\) ppm. Because the gross intensity of the Pb L\(_a\) peak for the dry wound model \(I_c(\ c = 0, t = 0, d = 66.0)\) is the sum of the net intensity of the Pb L\(_a\) peak and background intensity, it can be expressed as

\[
I_c(\ c = 0, t = 0, d) = I_{dc} + I_{od} \tag{1}
\]
According to the equation (1), both the slope and the intercept shown in Fig. 2c, $I_1 d$ and $I_0 d$, are directly proportional to $d$ (Fig. 3a and 3b). The parameters $I_1$ and $I_0$ can be, therefore, obtained as the slope of the datasets in Fig. 3a and 3b, respectively. Next, we focused on the correlation between the intensity of the XRF peak and blood thickness. Generally, the measured intensity, $i(t)$, transmitted through a layer of material with thickness, $t$, is related to the incident intensity, $i_0$, according to the Lambert law:

$$i(t) = i_0 e^{-\mu t}$$

(2)

where $\mu$ is called a linear attenuation coefficient. Because the net intensity, the first term in equation (1), obeys the Lambert law, the gross intensity of the Pb L\textalpha peak for the bleeding wound model $I(c, t, d)$ is expressed as

$$I(c, t, d) = I_1 e^{-\mu t} dc + I_0 d$$

(3)

The correlations between blood thickness and the net intensities of the Pb L\textalpha and Pb L\textbeta peaks relative to those for the dry wound model are shown in Fig. 3c. The linear attenuation coefficients of liquid blood at the energies of the Pb L\textalpha and Pb L\textbeta peaks can be obtained as the slopes of the datasets in Fig. 3c. As shown in Fig. 3c, the linear attenuation coefficient for the Pb L\textalpha peak is larger than that for the Pb L\textbeta peak. The ratios of the net intensities for the Pb L\textalpha peak to those for the Pb L\textbeta peak, L\textalpha/L\textbeta ratio, are plotted against the blood thickness (Fig. 3d). The data of the L\textalpha/L\textbeta ratio fit the straight line well in the semi-logarithmic plot, and this means the blood thickness, $t$, can be obtained from the observed L\textalpha/L\textbeta ratio. Using $I_1$, $I_0$ and $\mu$ values for the Pb L\textalpha peak obtained by least-square fittings (Fig. 3a, 3b and 3c), equation (3) can be rewritten as

$$I(c, t, d) = 1.15e^{-0.66t} + 1.74d.$$  

(4)

Here, blood thickness, $t$, can be obtained from the observed L\textalpha/L\textbeta ratio as described above. By least-square fitting the data in Fig. 3d, the L\textalpha/L\textbeta ratio, $R_i(t)$, is expressed as

$$R_i(t) = 0.87e^{-0.22t}.$$  

(5)

Thus, the blood thickness can be determined by applying the obtained L\textalpha/L\textbeta ratio from the measured XRF spectrum to equation (5).

Based on our findings, the following method for evaluating lead contamination in a wound is proposed. First, the XRF measurement is performed under the monitoring intensities of the Pb L\textalpha and Pb L\textbeta peaks. Second, the blood thickness is estimated by the L\textalpha/L\textbeta ratio obtained from the measured XRF spectrum and equation (5). Finally, the obtained blood thickness and maximal equivalent dose to the wound obtained from the accumulation time are assigned to equation (4) and the concentration of lead contamination in the wound can be evaluated.

Next, we focus on the minimum detection limit (MDL) of the measurement. Generally, MDL is defined as the concentration that gives the net intensity, $I_i e^{-\mu t} d$, three times the standard deviation of the background intensity. The MDL value can be, therefore, defined as the concentration which gives a signal three times the standard deviation of the background signal. The MDL value can be expressed as $3/\sqrt{\sigma d}$, because the standard deviation of the

...
background intensity is the square root of the background intensity. This value, $3\sqrt{I_0d}$, equals the net intensity when $c = \text{MDL}$ as

$$3\sqrt{I_0d} = I_1e^{-m}d\cdot\text{MDL}. \tag{6}$$

The MDL value is, therefore, expressed as

$$\text{MDL} = \frac{3\sqrt{I_0}}{I_1e^{-m}} \frac{1}{\sqrt{d}}. \tag{7}$$

As shown in equation (7), MDL is inversely proportional to the square root of the maximal equivalent dose to the wound. Fig. 4 shows the relation between maximal equivalent dose to the wound, $d$, and MDL ($d$-MDL curve) for each blood thickness ($t = 0, 0.5, 1.0, 1.5$ and $2.0$ mm) using experimentally defined parameters, $I_1$ (see Fig. 3a), $I_0$ (see Fig. 3b) and $m$ (see Fig. 3c). The $d$-MDL curve for any blood thickness can be also drawn. According to the $d$-MDL curve, MDL at any maximal equivalent dose to the wound can be calculated. In the present study, four blood phantoms with thicknesses of $t = 0.5, 1.0, 1.5$ and $2.0$ mm were prepared to consider the effect of blood thickness on the intensities of the signal. However, emergency first aid is given priority for patients with significant hemorrhage over evaluation of wound contaminants [23]. Thus, this method will be used on wounds with little hemorrhage, i.e., in which the blood exiting the wound is thin. For example, if the thickness of blood is $0.5$ mm and the maximal equivalent dose to the wound is $16.5$ mSv, which corresponds to $5$ s of accumulation time, MDL for lead can be calculated as approximately $1.2$ ppm. Considering the diameter of the measurement field for the portable XRF device used, the amount of substance in the measured region of the epoxy resin wound phantom containing $1.2$ ppm lead is approximately $3.1$ nmol. The corresponding radioactivity for the same amount of $^{239}$Pu is approximately $1.7$ kBq, which is lower than the radioactivity of $^{239}$Pu contaminating puncture wounds in previous severe accidents [24,25]. Therefore, this methodology could be applied to evaluate plutonium contamination in wounds and further tests on the feasibility are warranted. In the situation discussed above, the maximal equivalent dose to the area of a wound within the measurement field of the XRF device (diameter is $8$ mm) is estimated to be $16.5$ mSv. In a previous in vivo XRF measurement of patients [8], the equivalent dose to $1$-cm$^2$ area of skin was estimated as $13–26.5$ mSv, which is comparable to the maximal equivalent dose to the wound estimated in the present study. Incidentally, the equivalent dose limit of human skin is recommended as $500$ mSv annually for radiation workers by the International Commission on Radiological Protection (ICRP) [26]. Thus, the exposure dose in this method is acceptable.

For the evaluation of plutonium contamination in actual wounds in humans, the following modifications to the procedure used in the evaluation of lead contamination described above are recommended. Primarily, the treatment to stop the bleeding from the wound is carried out first. To examine the plutonium contamination, the following conditions are recommended. In the situation discussed above, the maximal equivalent dose to the area of a wound within the measurement field of the XRF device is estimated to be $16.5$ mSv. In a previous in vivo XRF measurement of patients [8], the equivalent dose to $1$-cm$^2$ area of skin was estimated as $13–26.5$ mSv, which is comparable to the maximal equivalent dose to the wound estimated in the present study. Incidentally, the equivalent dose limit of human skin is recommended as $500$ mSv annually for radiation workers by the International Commission on Radiological Protection (ICRP) [26]. Thus, the exposure dose in this method is acceptable.

For the evaluation of plutonium contamination in actual wounds in humans, the following modifications to the procedure used in the evaluation of lead contamination described above are recommended. Primarily, the treatment to stop the bleeding from the wound is carried out first. To examine the plutonium
contamination in wound, materials collected during the initial treatment, which include blood absorbed in the swab, are dried and subjected to a preliminary test by \( \gamma \)-survey meter. Although \( \gamma \)-survey meter determines the total amount of \( \gamma \)-particles emitted from several \( \gamma \)-emitters, the presence of plutonium is strongly indicated when \( \gamma \)-particles are counted. Under these circumstances, direct XRF measurement against the wound as described in this study is performed. For this analysis, the measuring time is initially adjusted to the maximum allowable time, which is calculated from the permissible radiation dose decided by clinician after considering the severity of the incident, predicted by the preliminary analysis. In a practical sense, establishment of guidelines for clinicians, which encourages rapid decision of measurement time, would be required. If there is concern for surface contamination on the skin surrounding the wound, masking the wound by a thin metallic sheet would be recommended. XRF analysis of the wound is then performed by monitoring the intensities of the Pu L\( \alpha \) and L\( \beta \) peaks. When the intensities of the peaks reach detectable levels for accurate determination of the concentration of plutonium, the measurement is stopped to ensure reduced radiation exposure. Finally, plutonium contamination in wound is quantified using equations (4) and (5), parameters for which are obtained from the measurement using the phantoms containing plutonium as described above. Such an accurate quantification can guide the subsequently employed therapeutic strategies. This method enables a patient to receive rapid treatment against contaminants (e.g., chelation therapy), which can reduce the internal exposure, although some additional skin doses would be given. Thus, the method described in this report can promote the use of the portable XRF device for real life emergencies.

This method rapidly evaluates the total plutonium content, but not the relative abundance of each isotope, which include \( ^{239}\text{Pu}, ^{240}\text{Pu}, \) and \( ^{241}\text{Pu}. \) When an actual accident occurs in the nuclear reactor, the relative amounts of the isotopes in the nuclear fuel splattered from reactor can be immediately estimated from the entries in the operating log. Using the total amount of plutonium contamination in the wound obtained by the described analysis and the estimated isotopic distribution derived from the operating log, the amount of each plutonium isotope in the wound can be calculated. Such an estimate is beneficial for the treatment of plutonium contamination. In addition, the wound can be contaminated with other radioactive elements (e.g., uranium and fission products (e.g., \( ^{90}\text{Sr}, ^{137}\text{I}, ^{134}\text{Cs}, \) and \( ^{137}\text{Cs} \)). However, these elements do not affect the background signal of the XRF peak for plutonium, since the XRF peak energy for plutonium is distinct and devoid of any overlap from the peaks attributed to those elements [15].

The XRF peak energies of Pu L\( \alpha \) (14.28 keV) and Pu L\( \beta \) (18.20 keV) are located in the region where intense background due to molybdenum peaks are observed. Therefore, we are developing a new device that uses only trace amounts of molybdenum near the path of incident X-ray. Additionally, to use this device for the analysis of a larger wound, our efforts are directed toward broadening the measurement range.

In this study, the epoxy resin wound phantoms constructed are 10-mm deep with a uniform distribution of contaminants to promote a conservative evaluation, which is accepted in radiation protection. A conservative evaluation is promoted for the following reason. In actual wounds, contaminants are considered to be usually located near the top of the wound. Since attenuation of X-rays increases with depth, the XRF intensity when contaminants are distributed at the surface of the wound is higher than that when the same amount of contaminant is uniformly distributed toward a 10 mm depth. Therefore, the amount of contaminants evaluated by the calibration curve obtained using a 10-mm-depth wound phantom is higher than the true value. Thus, the proposed method gives a conservative evaluation. Additionally, since the intensity of the XRF peak depends on the total amount of target atom, the quantitative evaluation of the plutonium contamination in wound can be determined even when the distribution of plutonium is laterally heterogeneous.

In conclusion, a methodology using a portable XRF device for on-site, rapid and direct evaluation of heavy-atom contamination in a wound was established using a bleeding wound model comprising an epoxy resin wound phantom containing contaminants and a blood phantom containing liquid blood. Developing the methodology using the liquid blood phantom was performed for the first time. The results of the present study indicate this methodology is applicable to evaluate plutonium contamination, which could contribute to the advancement of crisis management in the decommissioning of the Fukushima-Daiichi nuclear power plant. This method can also to be used for the analysis for contamination by actinides for injuries that occur in actinide handling facilities.

Acknowledgments

We thank Dr. S. Iwai (Japan Nuclear Technology Institute, Japan) for helpful discussion and members of the Atomic Energy Research Institute and Joint Research Center, Kinki University, Japan, for technical assistance.

Author Contributions

Conceived and designed the experiments: H. Yoshii. Performed the experiments: H. Yoshii HY NI MS. Analyzed the data: H. Yoshii KY YS. Contributed reagents/materials/analysis tools: H. Yoshii KY H. Imaseki TH. Wrote the paper: H. Yoshii NS OK KS.

References

1. National Council on Radiation Protection and Measurement (2006) Development of a biokinetic model for radionuclide-contaminated wounds and procedures for their assessment, dosimetry and treatment, NCRP report 156, Bethesda, MD: 118–133.
2. Saha GB (1993) Physics and radiobiology of nuclear medicine. New York: Springer. 11–20.
3. Genicot JL, Hardeman F, Oberstedt S (1995) The assessment of plutonium and americium in contaminated wounds with high energy resolution semiconductor detector. Appl Radiat Isot 46: 199–203.
4. Christoffersson JO, Mattsson S (1983) Polarised X-ray in XRF-analysis for improved in vivo detectability of cadmium in man. Phys Med Biol 28: 1135–1144.
5. Borjesson J, Barregård I, Sallsten G, Schütz A, Jonson R, et al. (1995) In vivo XRF analysis of mercury: the relation between concentrations in the kidney and the urine. Phys Med Biol 40: 413–426.
6. O’Meara JM, Chettle DR, McNell FE, Webber CE (1998) In vivo x-ray fluorescence (XRF) measurement of uranium in bone. Appl Radiat Isot 49: 713–715.
7. Nie H, Chettle D, Luo L, O’Meara J (2006) In vivo investigation of a new \( ^{109}\text{Cd} \) \( \gamma \)-ray induced K-XRF bone lead measurement system. Phys Med Biol 51: 351–360.
8. Nie LH, Sanchez S, Newton K, Grodzins L, Cleveland RO, et al. (2011) In vivo quantification of lead in bone with a portable x-ray fluorescence system—methodology and feasibility. Phys Med Biol 56: N39–N51.
9. Estevam M, Appoloni CR (2013) Use of portable X-ray fluorescence (PXRF) in vivo as an alternative technique for the assessment of iron levels in patients with thalassemia and hemochromatosis. Health Phys 104: 132–138.
10. O’Meara JM, Borjesson J, Chettle DR (2000) Improving the in vivo x-ray fluorescence (XRF) measurement of renal mercury. Appl Radiat Isot 53: 639–646.
11. Bradley DA, Farquharson MJ (2000) X-ray fluorescence and the in vivo evaluation of Fe, Cu and Zn in skin. J Radioanal Nucl Chem 244: 213–217.
12. Nie H, Chettle DR, McNeill FE, O’Meara JM (2004) An investigation of the 109Cd γ-ray induced K-x-ray fluorescence (XRF) bone-lead measurement calibration procedure. Phys Med Biol 49: N325–N334.
13. Fleming DE, Gherase MR (2007) A rapid, high sensitivity technique for measuring arsenic in skin phantoms using a portable x-ray tube and detector. Phys Med Biol 52: N439–N465.
14. Gherase MR, Vallee ME, Fleming DE. (2010) Simultaneous detection of As and Se in polyester resin skin phantoms. Appl Radiat Isot 68: 743–745.
15. Deslattes RD, Kessler EG Jr, Indelicato P, de Billy L, Lindroth E, et al. (2003) X-ray transition energies: new approach to a comprehensive evaluation. Rev Mod Phys 75: 33–99.
16. King LS (1942) Studies on eastern equine encephalomyelitis VI. Facilitation of infection in the mouse. J Exp Med 76: 325–334.
17. Trudnowski RJ, Rico RC (1974) Specific gravity of blood and plasma at 4 and 37°C. Clin Chem 20: 615–616.
18. Helmer OM, Emerson CP Jr (1934) The iron content of the whole blood of normal individuals. J Biol Chem 104: 157–161.
19. Iyengar GV (1987) Reference values for the concentrations of As, Cd, Co, Cr, Cu, Fe, I, Hg, Ma, Mo, Ni, Pb, Se and Zn in selected human tissues and body fluids. Biol Trace Elem Res 12: 263–295.
20. Moreno MA, Marin C, Vinagre F, Oszaczuk P (1999) Trace element levels in whole blood samples from residents of the city Badajoz, Spain. Sci Total Environ 229: 209–215.
21. Teng WF, Sun WM, Shi LF, Hou DD, Liu H (2008) Effects of restraint stress on iron, zinc, calcium, and magnesium whole blood levels in mice. Biol Trace Elem Res 121: 243–248.
22. Jenkins R (1999) X-ray fluorescence spectrometry. New York, John Wiley & Sons. 153–159 and 173–188.
23. International Atomic Energy Agency (2005) Serious medical problems always have priority over radiological concerns, EPR-MEDICAL. Vienna: IAEA. p 53.
24. Carbaugh EH, Lynch TP, Antonio CL, Valle FMD (2010) Twenty-four of follow-up for a Hanford plutonium wound case. Health Physics 99: 483–494.
25. Schadilov AE, Belosokov MV, Levina ES (2010) A case of wound intake of plutonium isotopes and 241Am in a human: application and improvement of the NCRP wound model. Health Physics 99: 560–567.
26. International Commission on Radiological Protection (2007) The 2007 Recommendations of the International Commission on Radiological Protection, ICRP Publication 103, Ann ICRP 37(2–4). New York: Elsevier. 98–106.