Cobalt-centred boron molecular drums with the highest coordination number in the CoB_{16} cluster

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The electron deficiency and strong bonding capacity of boron have led to a vast variety of molecular structures in chemistry and materials science. Here we report the observation of highly symmetric cobalt-centered boron drum-like structures of CoB_{16}, characterized by photoelectron spectroscopy and \textit{ab initio} calculations. The photoelectron spectra display a relatively simple spectral pattern, suggesting a high symmetry structure. Two nearly degenerate isomers with D\textsubscript{8d} (I) and C\textsubscript{4v} (II) symmetries are found computationally to compete for the global minimum. These drum-like structures consist of two B\textsubscript{8} rings sandwiching a cobalt atom, which has the highest coordination number known heretofore in chemistry. We show that doping of boron clusters with a transition metal atom induces an earlier two-dimensional to three-dimensional structural transition. The CoB_{16} cluster is tested as a building block in a triple-decker sandwich, suggesting a promising route for its realization in the solid state.
Boron, the fifth element in the periodic table, possesses such diverse chemical structures and bonding that are second only to carbon. Bulk boron consists of connected three-dimensional (3D) cages in many of its allotropes and boron-rich borides. However, for isolated clusters it was computationally shown that icosahedral cage structures of \(^{12}{\text{Bi}}_{12}\) and \(^{13}{\text{Bi}}_{13}\) were unstable, even though they were initially proposed as possible candidates for these two clusters. Over the past decade, small anionic boron clusters have been systematically characterized both experimentally and theoretically to exhibit planar or quasi-planar structures in their ground states up to \(^{17}{\text{B}}_{17}\) (refs 8–10). Recent works show that anionic boron clusters continue to be two-dimensional (2D) at \(^{18}{\text{Bi}}_{30}\) (ref. 11), \(^{19}{\text{Bi}}_{40}\) (ref. 12) and \(^{20}{\text{Bi}}_{46}\) (ref. 13). The 2D-to-3D transition was suggested to occur at \(^{39}{\text{B}}_{20}\) for neutral, and at \(^{40}{\text{Bi}}_{29}\) for cationic clusters. Very recently it is shown that the transition from 2D to fullerene-like 3D structures occurs in negatively charged boron clusters at about 40 boron atoms in \(^{18}{\text{Bi}}_{30}\) (ref. 16) and \(^{20}{\text{Bi}}_{36}\) (ref. 17). Due to the nearly spherical shapes of these clusters, they have been named borospherenes. Doping boron clusters with a single metal atom opens a new avenue to create clusters with novel structures and chemical bonding. It has been experimentally observed that various transition metal atoms can be placed inside of monocyclic boron rings to form beautiful molecular wheel-type structures (M\(\subset^{12}{\text{B}}_{12}\))\(^{1}\), following an electronic design principle inspired by the doubly \(\sigma\) and \(\pi\) aromatic \(^{12}{\text{B}}_{12}\) cluster.\(^{18}\) It was shown that the Nb\(\subset^{12}{\text{B}}_{10}\) and Ta\(\subset^{12}{\text{B}}_{10}\) clusters possess the record coordination number of 10 in the planar environment for the central metal atom.\(^{20}\) These clusters have pushed the limits of structural chemistry.

Here we report the observation of a large metal-doped boron cluster of \(^{14}{\text{B}}_{14}\), which is produced using a laser vaporization cluster source and characterized by photoelectron spectroscopy (PES). Extensive computational searches reveal that there are two nearly degenerate structures for \(^{14}{\text{B}}_{14}\), which are indistinguishable at the highest level of theory employed. They both possess tubular double-ring framework and give similar photoelectron spectral patterns. The structures can be viewed as two \(^{12}{\text{B}}_{8}\) rings sandwiching a Co atom, reminiscent of a drum and giving rise to the highest coordination number known in chemistry thus far.

Results

Experimental results. The photoelectron spectra of \(^{14}{\text{B}}_{14}\) at two photon energies are displayed in Fig. 1. The lowest binding energy band (X) represents the electron detachment transition from the anionic ground state to that of neutral \(^{14}{\text{B}}_{14}\). The higher binding energy bands, A, B, ..., denote detachment transitions to the excited states of neutral \(^{14}{\text{B}}_{14}\). The vertical detachment energies (VDEs) for all observed bands are given in Table 1, where they are compared with the calculated VDEs.

The 266 nm spectrum (Fig. 1a) reveals three well-resolved PES bands for \(^{14}{\text{B}}_{14}\). The band X gives rise to a VDE of 2.71 eV. The adiabatic detachment energy (ADE) for band X was evaluated from its onset to be 2.48 eV, which also represents the electron affinity of neutral \(^{14}{\text{B}}_{14}\). The width of band X suggests an appreciable geometry change between the electronic ground state of \(^{14}{\text{B}}_{14}\) and the ground electronic state of \(^{14}{\text{B}}_{14}\). Following a relatively large energy gap, an intense and broad band A is observed at a VDE of 3.45 eV and a close-lying band B at a VDE of 3.78 eV. The 193 nm spectrum (Fig. 1b) shows nearly continuous signals beyond 4 eV. The sharp spikes above 5 eV in the high binding energy side of the 193 nm spectrum are due to statistical noises because of low electron counts. An intense and broad band C is clearly observed at a VDE of 4.86 eV. Two more bands can be tentatively identified at higher binding energies, D (VDE: ~5.3 eV) and E (VDE: ~5.6 eV). Overall, the PES spectral pattern is relatively simple, suggesting that the framework of the \(^{14}{\text{B}}_{14}\) cluster is likely to have high symmetry.

Theoretical results and comparison with experiment. Extensive structural searches were initially done at the PBE0/3-21G level of theory with the follow-up calculations (\(\Lambda = 25\) kcal mol\(^{-1}\)) at the PBE0/Def2-TZVP level of theory, which led to two similar drum-like structures: isomer I (\(^{12}{\text{Nb}}_{6}\), \(^{3}\text{A}_{2}\)) and isomer II (\(^{12}{\text{Ta}}_{6}\), \(^{1}\text{A}_{1}\)) identified as the global minima for \(^{14}{\text{B}}_{14}\) (Fig. 2). These two highly symmetric structures, consisting of a central Co atom sandwiched by two \(^{12}{\text{B}}_{8}\) monocyclic rings, are found to be almost degenerate at various levels of theory (Supplementary Fig. 1 and Supplementary Table 1). Clearly, the method dependency of predicting relative energies of the low-lying structures for \(^{14}{\text{B}}_{14}\) suggests the importance of comparison with experiment in determining the global minimum. We previously studied how optimized geometries of small boron clusters differed at density functional theory (DFT) and CCSD(T) levels of theory. We found that B3LYP is adequate for \(6-311+\text{G}^\ast\) geometry optimizations but that single-point energy calculations were performed at ROCCSD(T)/6-311+\(\text{G}^\ast\) using optimized UPBEO/Def2-TZVP geometries here and elsewhere. These small values are in the range of the theoretical errors for such a complex transition-metal-doped boron cluster. Therefore, isomers I and II should be considered to be degenerate based on our calculations.

Figure 2 shows the small differences in bond distances between isomers I and II; the latter is not significantly distorted from the \(^{12}{\text{B}}_{8}\) symmetry. The B–B bond lengths of the \(^{12}{\text{B}}_{8}\) rings for both isomers are in the range of 1.55–1.63 Å, similar to the corresponding values (1.56 Å) in the \(^{12}{\text{Co}}_{10}^{\text{B}}\) molecular wheel. The nearest isomer III (\(^{12}{\text{Co}}_{8}\), \(^{1}\text{A}_{1}\)) is 8.7 kcal mol\(^{-1}\) higher in energy at the ROCCSD(T) method and represents a
Table 1 | Experimental and theoretical vertical electron detachment energies (VDEs) in eV of CoB₁₆.

| MO   | VDE (exp.) | Isomer I | Isomer II |
|------|------------|----------|-----------|
|      |            | (1a₁²1e₂¹1e₂¹1e₂²1e₂²2a₂²2a₂²2e₂²2e₂²) | (1a₁²1e₂¹1e₂¹1e₂²1e₂²2a₂²2a₂²2e₂²2e₂²) |
|      |            | 3a₁²3e₂¹2e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂²3e₂² |
|      |            | 4e₂² | 2.24 | 4e₂² | 2.24 |
|      |            | 2b₂² | 2.91 | 2b₂² | 2.91 |
|      |            | 4e₂² | 2.59 | 4e₂² | 2.59 |
|      |            | 4b₂² | 2.53 | 4b₂² | 2.53 |
|      |            | 4b₂² | 2.47 | 4b₂² | 2.47 |
|      |            | 7a₁² | 3.02 | 7a₁² | 3.02 |
|      |            | 7a₁² | 3.09 | 7a₁² | 3.09 |

*Numbers in parentheses indicate the uncertainties of the last digit. The ADE of the X band is measured to be 2.48(5) eV.
†The VDEs were calculated at the UB3LYP/6-311+G(2df)/UB3LYP/Def2-TZVP level of theory. Spin contamination was found to be very small.
‡The VDEs were calculated at the ROCCSD(T)/6-311+G(2df)/ROCCSD(T)/Def2-TZVP level of theory because the UHF wave function has a very high spin-contamination.
\(^{3}\)VDE could not be calculated at this level of theory.

Figure 2 | Two views of isomer I and isomer II of the CoB₁₆ cluster. The point group symmetries and spectroscopic states of isomer I (a) and isomer II (b) are shown in parentheses. Sticks drawn between atoms help visualization and do not necessarily represent classical Zn–Zn B–B or Co–B bonds here and elsewhere. All distances are in Å.

distorted drum-like structure composed of two B₇ rings with two B atoms outside the drum (Supplementary Fig. 1). In fact, the majority of the low-lying isomers within 20 kcal mol⁻¹ (Supplementary Fig. 1) represent various derivatives (drum-like or possessing principal geometrical features of the drum-like structure) of isomers I and II, showing the stability of the drum-like structures. It should be noted that there are significant bonding interactions between the two B₇ rings and between the Co atom and all 16 B atoms in both isomers I and II (vide infra).

Interestingly, the drum structure in a quintet state (isomer XIV in Supplementary Fig. 1) appears to be the most stable one out of all other quintet isomers. It should be mentioned that there were two previous DFT calculations on similar drum-like structures of neutral boron clusters doped with transition metal atoms²⁴,²⁵.

To facilitate comparisons between the experimental and theoretical results, we calculated low-lying VDEs of isomers I and II of CoB₁₆ using three methods (Table 1). We found that the VDEs computed using the two DFT methods are not very impressive; but we observed good agreement between the theoretical VDEs at ROCCSD(T)/6-311+G(2df) and the experimental data for the first two detachment channels (Table 1). Since isomer I is open shell, the electron detachment energy from the doubly degenerate 4e₂¹–HOMO should lead to a doublet final state for the neutral. The computed VDE at ROCCSD(T) is 2.59 eV, compared with the experimental VDE of 2.71 eV. The next electron detachment from the non-degenerate 2b₂²–HOMO-1 should lead to both a quartet and a doublet final state, with the quartet being lower in energy. The calculated VDE for the quartet final state at ROCCSD(T) is 3.28 eV, compared with the VDE of the A band at 3.45 eV. Unfortunately, we were not able to calculate any higher VDEs because of the limitation of the ROCCSD(T) method. However, we believe that the good agreement between experiment and theory for the first two VDEs provides sufficient credence for the identified drum-like isomer I for the CoB₁₆ cluster.

Isomer II gives very similar theoretical VDEs as isomer I at all three levels of theory, consistent with the similarities in their geometries. Since isomer II is a closed shell species, we were able to calculate only the first VDE value at the ROCCSD(T) method as 2.61 eV, also in good agreement with the experimental data. Furthermore, the calculated ADEs of isomer I (2.45 eV) and isomer II (2.43 eV) (PBE0/Def2-TZVP) are in excellent agreement with the experimentally measured ADE value of 2.48 eV. We should point out that there is a Jahn–Teller distortion for the neutral CoB₁₆ drum-like structure of isomer I, consistent with the broad X band observed in the PES spectra (Fig. 1). Indeed, the calculated relaxed neutral CoB₁₆ structure I⁰ (Supplementary Fig. 2 and Supplementary Table 1) has lower symmetry (C₃ᵥ), as one would expect for the Jahn–Teller distorted structure due to the occupation of the doubly degenerate HOMO (4e₂) of isomer I by a single electron. In fact, the HOMO (4b₂) of isomer II originates from the HOMO (4e₂) of isomer I when one of the doubly degenerate orbitals is doubly occupied. Therefore, the detachment of one electron from the doubly occupied HOMO (4b₂) of isomer II leads to the same neutral structure I⁰. The high relative energy of isomer III, as well as its appreciably higher theoretical first VDE of 3.65 eV (Supplementary Table 2), makes this cluster unlikely to be populated in the molecular beam in any appreciable amount.

Discussion

Tubular (or drum-like) boron clusters have been of interest for many years, because they can be considered as the embryos for boron nanotubes¹⁴. However, such drum-like structures have never
been observed experimentally for bare boron clusters, even though they have been shown to be stable computationally\(^1\),\(^2\),\(^3\),\(^4\),\(^5\),\(^6\). For instance, the B\(_{3\alpha}\) cluster was first suggested as the global minimum on the basis of theoretical calculations\(^7\),\(^8\), but it has not been observed or confirmed experimentally\(^9\). Tubular structures were also studied for the bare B\(_{6\alpha}\), B\(_{16\alpha}\) and B\(_{16\beta}\) species\(^\text{3,5}\). For the B\(_{3\alpha}\) cationic cluster, the tubular structure was suggested to be the global minimum\(^1\), whereas the tubular structures of both B\(_{4\alpha}\) and B\(_{5\alpha}\) were found to be high-energy isomers\(^5\). Clearly, the strong coordination interactions with the Co atom significantly stabilize the tubular B\(_{n}\), giving the drum-like global minima (structures I and II) for CoB\(_{1\alpha}\). Bare anionic boron clusters are found to be 2D up to B\(_{8\alpha}\) (ref. 13), while some transition-metal-doped anionic boron clusters are found to preserve the planar boron framework on metal doping\(^1\),\(^2\),\(^3\). The largest experimentally observed metal-doped boron cluster (CoB\(_{1\alpha}\)) maintains a similar planar geometry for the B\(_{7}\) moiety\(^5\). Hence, the doping of the Co atom induces an earlier 2D-to-3D transition for boron clusters, as shown by the 3D isomers I and II of CoB\(_{1\alpha}\). In fact, the CoB\(_{1\alpha}\) drum structure represents the highest coordination number known in chemistry today. The previous highest coordination number known experimentally was 15 for [Th(H\(_3\)BNMe\(_2\)BH\(_3\))]\(_n\) (ref. 33), though theoretical studies have assessed to be around 0.5 |e|. The last set includes five delocalized bonds, which represent π–π interactions between the boron rings: three 16c–2e π–π bonds (Fig. 3k–m) with ON = 1.98–2.00 |e| and two 16c–1e π–π bonds (Fig. 3n,o) with ON = 1.00 |e| (one unpaired electron on each bond). The eight π electrons in the last set suggest π-aromaticity according to the 4n rule (n = 2) for triplet states. Therefore, the stability of isomer I of CoB\(_{1\alpha}\) can be considered to be due to the double π- and π-aromaticity and bonding interactions of the 3d AOs of Co with the B\(_{8}\) rings.

As expected, isomer II of CoB\(_{1\alpha}\), which is close in energy and geometry to isomer I, has almost the same bonding pattern as that of isomer I (Supplementary Fig. 3). All the bonding elements found in isomer I are also found in isomer II except for the last set (Supplementary Fig. 3k–n). Since isomer II is closed shell, eight electrons in the last set are observed to form four 16c–e π–π bonds with ON = 1.98–2.00 |e|, rendering this isomer π-antiaromatic. Hence, isomer II exhibits conflicting aromaticity (π-aromatic and π-antiaromatic), which leads to some distortion to C\(_{6v}\) symmetry compared to the D\(_{3d}\) symmetry of the doubly aromatic isomer I. As was mentioned earlier, the HOMO (4b\(_1\)) of isomer II originates from the HOMO (4e\(_2\)) of isomer I when one of the doubly degenerate orbitals is doubly occupied. Indeed, occupation of only one degenerate MO by two electrons causes the electronic instability, which causes the geometric rearrangement of isomer II lowering the D\(_{3d}\) symmetry to C\(_{6v}\).

To understand the interactions between Co and the tubular B\(_{16}\) host, we have performed AdNDP analyses for the neutral B\(_{16}\) tubular isomer (Supplementary Fig. 4). Similar to isomers I and II of CoB\(_{1\alpha}\), the AdNDP analyses give 16 2c–2e B–B σ-bonds with ON values of 1.70 |e| within the two B\(_{8}\) rings. The encapsulation of Co strengthens the B–B σ-bonds within each B\(_{8}\) ring in CoB\(_{1\alpha}\), but weakens the inter-ring interactions, compared with the bare B\(_{16}\) as reflected by their ON values (Fig. 3 and Supplementary Fig. 4) and the B–B bond lengths (Supplementary Table 3). The remaining 16 electrons in B\(_{16}\) participate in delocalized bonding: five 16c–2e σ + σ bonds and three 16c–2e π–π bonds, rendering the tubular B\(_{16}\) doubly σ- and π-aromatic. The major difference in chemical bonding between the drum-like B\(_{16}\) and CoB\(_{1\alpha}\) comes from two factors: (1) the formation of an additional set (Fig. 3h–j) of the delocalized σ–σ bonds in CoB\(_{1\alpha}\) and (2) participation of Co 3d AOs in the two 17c–2e σ + σ bonds (Fig. 3f,g). Both factors are consistent with structural changes between CoB\(_{1\alpha}\) and B\(_{16}\). There are strong bonding interactions between Co and the B\(_{16}\) host in CoB\(_{1\alpha}\) to stabilize the tubular B\(_{16}\) structure, because the global minimum of B\(_{16}\) is planar\(^3\).

Isomer I of CoB\(_{1\alpha}\) is open shell with two unpaired electrons, whereas isomer II can be viewed as a result of Jahn–Teller distortion from isomer I. Addition of two electrons to isomers I or II would create a closed shell and doubly aromatic CoB\(_{1\alpha}\)\(_{1\beta}\) species with D\(_{3d}\) symmetry. Our calculations indeed confirmed this hypothesis: CoB\(_{1\alpha}\)\(_{1\beta}\) was found to be a minimum on the potential energy surface with very similar bond distances as in isomer I (Supplementary Table 3). The triply charged CoB\(_{1\alpha}\)\(_{1\beta}\) species can be electronically stabilized by external akali metal...
cations, such as in Na₂CoB₁₆. Since ligation would be needed to ultimately synthesize CoB₁₆, we considered a triple-decked [CoB₁₆(CaCp)₂]⁻ sandwich complex (Supplementary Fig. 5), using the divalent Ca atoms and the aromatic C₅H₅⁻ (Cp⁻) ligands. It should be mentioned that similar [CpLiB₆LiCp]₂⁻ triple-decked complex⁴² with the double anti-aromatic B₆²⁻ unit was previously suggested to be stable and viable experimentally. We found that the [CoB₁₆(CaCp)₂]⁻ triple-decked complex was a minimum on the potential energy surface with high electronic stability. All the B–B and Co–B bond lengths were found to be almost the same as in isomers I and II of CoB₁₆ (Supplementary Table 3). We have further performed AdNDP analyses and found that the triple-decked sandwich complex exhibits exactly the same chemical bonding pattern as the parent CoB₁₆ (Supplementary Figs 6–8). The Natural Population Analysis (NPA) charge on Ca was found to be +1.54, consistent with the initial hypothesis and the charge-transfer nature of the triple-decked [CoB₁₆(CaCp)₂]⁻ sandwich complex. Thus, the CoB₁₆ molecular drum can serve as a building block for the design of novel cluster-assembled nanomaterials. The high stability of the CoB₁₆ drum structures may also help the search for new metal-boride phases containing various boron ring units⁴³.

We have produced and characterized a large Co-doped boron cluster, CoB₁₆, using photoelectron spectroscopy and quantum-chemical calculations. Extensive computational searches established two high symmetry (D₈d and C₄ᵥ) drum-like structures with Co sandwiched by two B₈ rings as nearly degenerate global minima. The CoB₁₆ molecular drums represent the highest coordination for a metal atom known in chemistry and opens new possibilities for designing novel boron-based nanomaterials. First, the CoB₁₆ drums may be considered as the embryo to make filled boron nanotubes due to the significant B–B bonding between the two B₈ rings. Second, there are possibilities to

Figure 3 | Chemical bonding picture. (a–o) The overall chemical bonding picture (a–o) obtained for the isomer I of the CoB₁₆ molecular drum via the UAdNDP analysis. ON denotes occupation number here and elsewhere.
observe larger doped-boron clusters with even higher coordination number to further push the limit of coordination number in chemistry. Third, we have demonstrated one possibility to use CoB16 as a building block of new cluster-assembled nanomaterials in a triple-decked complex.

Methods

Experimental methods. The experiment was carried out using a magnetic-bottle PES apparatus equipped with a laser vaporization cluster source44. Briefly, the CoB16 clusters were produced by laser vaporization of a cold-target composed of Co and isotopically enriched11B. Bismuth was added as a binder and it also provided a convenient calibrant (Bi+) for the PES experiment. Clusters formed in the nozzle were entrained in a He carrier gas and underwent a supersonic expansion to form a collimated cluster beam. The He carrier gas was seeded with 5% Ar for better cooling of the entrained clusters52. The anionic clusters were extracted from the collimated cluster beam and analysed by a time-of-flight mass spectrometer. The CoB16 anion clusters were mass selected and decelerated before being photodetached by a laser beam at 193 nm (6.424 eV) from an ArF excimer laser or 266 nm (4.661 eV) from a Nd:YAG laser. Photodecays were carried out in a 3.5 m long flight tube. The resolution of the apparatus, AEK/Ek, was about than 2.5%, that is, ~25 mV/eV electrons.

Theoretical methods. Search for the global minimum of CoB16 was performed using the Coalescence Kick algorithm45 at the PBE0/6-31G level of theory46,47. The Coalescence Kick algorithm generated ~10,000 trial structures for each spin multiplicity (singlet, triplet and quartet), followed by geometry optimization. Low-lying isomers within 25 kcal mol−1 were further refined at a more expansive basis set, Def2-TZVP48. For each structure, vibrational frequencies were calculated and imaginary frequencies were followed to ensure that the isomer corresponded to a true minimum on the potential energy surface. Spin contamination was found to be <10% in all DFT calculations. For selected isomers, we performed additional geometry optimization at various DFT levels, as well as more accurate single-point coupled-cluster calculations [ROCCSD(T)/6-31+G(2d,2p)], to reliably establish the relative energy ordering. Vertical detachment energies of the three lowest energy structures were calculated at three different methods (UPBE0, UB3LYP and ROCCSD(T)) to compare with the experimental data. The VDEs were obtained as the difference in energy between the ground state of the anion and selected low-lying electronic states of the neutral molecule at the geometry of the anion. All calculations were done using GAUSSIAN-09 (ref. 49).

To understand the chemical bonding, we carried out electron localization analyses using the AdNDP method50 at the PBE0/6-31G(d) level of theory. Previously, AdNDP results have shown be insensitive to the level of theory or basis set used51,52. The AdNDP analysis is based on the concept of electron pairs as a building block of new cluster-assembled nanomaterials in a triple-decked complex.

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**Author contributions**

L.S.W. and A.I.B. designed the research. I.A.P. and A.I.B performed and analyzed the calculations. L.S.W., T.J. and G.V.L. designed experiments and analysed the experimental data. All authors contributed to the interpretation and discussion of the data. I.A.P. and T.J. wrote the manuscript.

**Additional information**

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