Reduction of metastasis using a non-volatile buffer

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Abstract The tumor microenvironment is acidic as a consequence of upregulated glycolysis and poor perfusion and this acidity, in turn, promotes invasion and metastasis. We have recently demonstrated that chronic consumption of sodium bicarbonate increased tumor pH and reduced spontaneous and experimental metastases. This occurred without affecting systemic pH, which was compensated. Additionally, these prior data did not rule out the possibility that bicarbonate was working through effects on carbonic anhydrase, and not as a buffer per se. Here, we present evidence that chronic ingestion of a non-volatile buffer, 2-imidazole-1-yl-3-ethoxycarbonylpropionic acid (IEPA) with a pKa of 6.9 also reduced metastasis in an experimental PC3M prostate cancer mouse model. Animals (n = 30) were injected with luciferase expressing PC3M prostate cancer cells either subcutaneously (s.c., n = 10) or intravenously (i.v., n = 20). Four days prior to inoculations, half of the animals for each experiment were provided drinking water containing 200 mM IEPA buffer. Animals were imaged weekly to follow metastasis, and these data showed that animals treated with IEPA had significantly fewer experimental lung metastasis compared to control groups (P < 0.04). Consistent with prior work, the pH of treated tumors was elevated compared to controls. IEPA is observable by in vivo magnetic resonance spectroscopy and this was used to measure the presence of IEPA in the bladder, confirming that it was orally available. The results of this study indicate that metastasis can be reduced by non-volatile buffers as well as bicarbonate and thus the effect appears to be due to pH buffering per se.

Keywords Reduction of metastasis · Buffering · pH · Acidic microenvironment · PC3M · Prostate cancer

Abbreviations

FSEMS Fast spin echo multi slice; a pulse sequence for use in T2-weighted magnetic resonance imaging

FOV Field of view used in magnetic resonance acquisition

IACUC Institutional animal care and use committee

IEPA 2-Imidazole-1-yl-3-ethoxycarbonylpropionic acid

MEM/EBSS Minimum essential media/Earle’s balanced salts

MMP Matrix metalloproteinase

MR/I/S Magnetic resonance imaging/spectroscopy

OD Outer diameter

PBS Phosphate buffered saline

STEAM Stimulated echo acquisition mode; a varian pulse sequence for localized magnetic resonance spectroscopy

TE/TR/TM Echo time/repetition time/mixing time

MR Pulse sequence parameters
Introduction

Acidosis is a common microenvironmental feature of most solid tumors [1]. Acidification of the extra cellular space is a consequence of high rates of glucose metabolism combined with inefficient tumor perfusion [2, 3]. Low extra-cellular pH can induce tumor cell migration, invasion and metastasis by poorly defined mechanisms [4, 5]. Acute exposure of cells to acidic pH has been shown to cause up-regulation of several secreted proteases, such as cathepsins D, L, and/or B, [6–8] and increase expression of the matrix metalloproteinases (MMP); MMP-2 (gelatinase A) and MMP-9 (gelatinase B) in vitro [9, 10]. Additionally, pre-treatment of melanoma cells with acidic pH before tail vein injection leads to increased experimental metastases in vivo [11, 12]. We have shown previously that neutralizing the acid pH of tumors with oral NaHCO₃ reduced spontaneous metastases without affecting systemic pH [13]. Mathematical reaction-diffusion type models were used to quantitatively describe the effect, thus providing a theoretical framework within which to interpret the data. The pKₐ of NaHCO₃ in water is 6.14 [14]; and models predicted that the most efficacious buffers would have a pKₐ of ~ 7.0. Additionally, the models assumed that bicarbonate was working as a buffer per se, and did not assume a specific interaction with carbonic anhydrases, which are used in vivo to dehydrate HCO₃⁻. The current study was undertaken to test the efficacy of a non-bicarbonate/non-volatile buffer with a higher pKₐ on tumor growth and metastases. For this, we have used a non-volatile buffer with a pKₐ of ~ 6.9, 2-imidazole-1-yl-3-ethoxycarbonyl-propionic acid (IEPA).

IEPA is a synthetic imidazole buffer that has been used to map the extracellular pH, pHe, in different animal models using magnetic resonance spectroscopic imaging, MRSI [15–17]. In this paper, we studied the effect of IEPA on spontaneous and experimental metastases. We first showed that oral IEPA was tolerable at doses up to 200 mM, as evidenced by animal weight gain. We demonstrated that IEPA was absorbed into the systemic circulation using MRS. Measurement of the pH in the primary subcutaneous tumor showed that IEPA slightly increased the tumor pH. IEPA also had a moderate but statistically insignificant effect on the growth rate of the primary tumor, which was consistent with previous results using bicarbonate. We then observed that IEPA had a significant inhibitory effect on experimental metastases in the PC3M prostate model. Although the mechanism by which inhibiting metastases has not yet been determined, these data combined with the previous bicarbonate studies [14] demonstrate that non-volatile buffers are effective in reducing tumor metastases, and thus the effects appear to be due to buffering per se.

Materials and methods

Chemicals

IEPA (2-imidazole-1-yl-3-ethoxycarbonyl-propionic acid) was obtained from Dr. Paloma Ballesteros (Laboratory of Organic Synthesis and Molecular Imaging by Magnetic Resonance (Ref. Lab. 250) Faculty of Sciences, UNED Pº Senda del Rey, 9 28040-Madrid, Spain), and was dissolved in tap water at a concentration of 200 mM. The pH of the IEPA solution was adjusted to 7.4 (using 1 M HCl) and then given to the mice to drink in place of their regular tap water. Cell culture media and supplies were obtained from Invitrogen, Carlsbad, CA. Luciferin was obtained from Gold Biotechnology, St. Louis, MO.

Animals housing and diet

All animals were maintained in accordance with IACUC standards of care in pathogen free rooms, in the Moffitt Cancer Research Center (Tampa, FL) Vivarium. All imaging and measurements were performed within the facility. Three days prior to inoculation with tumor cells, 4–6 week old male beige SCID mice (Harlan, Madison, WI) were placed in two cohorts per experiment, which were allowed to drink either tap water or 200 mM IEPA. The weights of the water bottles were recorded before and after providing them to the animals, thereby tracking the amount of liquid consumed per cage. Animal weights were measured and recorded twice weekly, and the overall health of each animal was noted to ensure timely endpoints within the experiment.

Cell culture and inoculation

PC3M cells (-Luc6 clone) were obtained from Caliper (Hopkinton, MA). The cells were cultured using MEM/EBSS media, supplemented with 10% fetal bovine serum, 1% penicillin streptomycin, 1% nonessential amino acids, 1% sodium pyruvate and 1% MEM vitamins. In preparation for inoculation into mice, the cells were trypsinized and rinsed once with sterile phosphate buffered saline (PBS) prior to resuspension at a concentration of 5 × 10⁶ cells in 200 μl PBS. For primary tumor injection, animals were prepared by removing the hair from the injection site, and 200 μl containing 5 × 10⁶ cells in PBS were injected subcutaneously into the right flank of each mouse. For experimental metastases, 200 μl containing 5 × 10⁶ cells in PBS were injected directly and slowly (over the course of 1 min) into the tail vein of each mouse. In both preparations, cell distributions were verified by bioluminescent imaging immediately following injection.
Bioluminescent imaging

Animals were anesthetized with isoflurane and injected intraperitoneally with 10 μL per g body weight of sterile d-luciferin substrate prepared in PBS at 15 mg/ml (resulting dose 150 μg/g body weight). After 5 min, mice were transferred to the thermo-regulated, light-tight chamber of the In Vivo Imaging System, IVIS-200 (Caliper; Hopkinton, MA). Photographic images were acquired first, and the bioluminescent images were overlaid on top of these images. Bioluminescent images were acquired by measuring photons emitted from luciferase-expressing cells and transmitted through the tissue. The exposure time for the bioluminescent image acquisition ranged from 0.5 s (whole tumor images) up to 2 min (lung metastases) to ensure non-saturation, and differences in exposure time were corrected by expressing data as total flux in photons/sec, rather than photon counts. Images were analyzed using the Living-Image software (Caliper; Hopkinton, MA).

Necrosis counting

The center section (~5 mm) of the subcutaneous tumor of each animal was fixed in paraffin blocks prior to staining one 4 μm thick cross-sectional sample per animal with hematoxylin and eosin for histology. Histology slides were scanned using the Aperio™ (Vista, CA) ScanScope XT with a 20×/0.8NA objective lens (200×) at a rate of 2 min per slide via Basler tri-linear-array. Image analysis was performed using an Aperio Genie® v1 customized algorithm in conjunction with Positive Pixel Count v9 with the following optimized thresholds: Hue value = 0.2; Hue width = 0.6; color saturation threshold = 0.05; Iwp(High) = 210; Ip(Low) = Ip(High) = 160; Ip(High) = Ip(Low) = Ip(High) = 80; Ip(Low) = 0]. The algorithm was applied to the entire slide’s digital image to determine the percentage of necrosis by detecting the number of pixels that satisfy the color and intensity specification defined above (necrotic), divided by the number of pixels in non-necrotic tissue. The training algorithm developed above was quality controlled by a practicing pathologist.

Magnetic resonance imaging and spectroscopy

MR images and spectra were obtained on a Varian MR imaging spectrometer ASR310 (Agilent Life Sciences Technologies, Santa Clara, CA) with a 30 cm horizontal clear bore operating at a field strength of 7 T. For reference, a high resolution spectrum of IEPA in D2O was obtained on a Varian Nuclear Magnetic Resonance spectrometer with a 54 mm vertical bore opening at a field strength of 9.4 T. For in vivo spectroscopic imaging, naïve mice were allowed to drink IEPA for 3 days prior to imaging. The animals were sedated using isoflurane, placed in the animal cradle for insertion into the bore of the 7 T Varian MRI and maintained warm using a continuous warm air blower (Small Animal Instruments, Inc., Stonybrook, NY). Temperatures were measured using a fiber optic endorectal thermometer in conjunction with the MR compatible animal monitoring system (Model 1025, Small Animal Instruments, Inc. Stony Brook, NY.). SCOUT images were taken to verify location, and T2 weighted images for anatomical identification were obtained using an fast spin echo (FSEMS) pulse sequence, with FOV = 40 × 80 (mm), 15 coronal slices, 1 mm thick, no gap, TR = 2450 s, effective TE = 72 ms, with fat suppression on. Spectra were obtained using a stimulated echo (STEAM) localization sequence on a 2 × 2 × 2.5 mm3 voxel in the bladder with 256 averages (flip angle 90 deg, TE 9.44, T1 8.01, and TR 2000), for an 8.5 min acquisition. Images and spectra were processed using the Varian Vnmrj software or using MATLAB (MathWorks, Inc, Natick, MA).

Electrode measurement of pH

Animals were sedated using isoflurane, and placed on a warming surface to maintain appropriate body temperature for the duration of the experiment. Both the needle microelectrode and the reference electrode were obtained from Microelectrodes, Inc., (Bedford, NH). A shallow small (<5 mm) incision was made in an alternate (non tumor) site and the 1 mm reference electrode was placed subcutaneously therein. A needle micro electrode (OD 0.8 mm with a beveled end) was inserted up to 1.3 cm into the center of the tumor, and was held in place for up to 1 min, until pH readings stabilized. The needle was rotated once in each location, to allow the pH electrode to re-read at the same depth in order to make two independent measurements per location. The pH was measured at three locations, one near the center/core of the tumor, one in a mid region of the tumor, and one at the rim of the tumor; these values were averaged to report a mean for each animal. After the pH was measured in the primary tumor, the animals were euthanized (29 days after subcutaneous injection of the primary tumor cells). Before and after the pH was measured in each animal, the pH electrodes were used to measure a standard pH 7 buffer solution (Thermo Fisher Scientific, Inc., Waltham, MA).

Statistics

Non-paired data were analyzed for differences in their mean by two-tailed Student’s t-test. Chauvenet’s criterion was used for outlier analysis of independent data sets. A mouse was rejected from its set based on an expected deviance value of below one-half. The expected deviance
values for all other physical measurements were above one-half showing two otherwise valid sets [18].

Results

IEPA is tolerable and orally available

IEPA (Fig. 1a) is a synthetic imidazolic base with a $pK_a$ of 6.9. When added to drinking water at a concentration of 200 mM, mice appear to thrive. The water consumption was not significantly different in the IEPA group compared to that of animals consuming tap water (Fig. 1b). No behavioral alterations were noted and there was no effect on animal weight maintenance or gain (Fig. 1c).

IEPA can be detected in vivo via magnetic resonance spectroscopy, which has been used as a pH sensitive agent using magnetic resonance spectroscopic imaging, MRSI [17]. Although IEPA is an MRS imaging agent, a pH electrode was chosen for this experiment in order to avert potential confounding effects of this buffer on the measurement itself. To image pH with IEPA, it must be given

Fig. 1 a Name and chemical structure of IEPA. b Average water consumption over time in the two cohorts (Tap and IEPA; $n = 10$ each) indicating no difference in the amount of water consumption between the treatments. c Average mouse weight over time between cohorts of mice drinking either tap water (Tap) or tap water buffered with 200 mM IEPA (IEPA), showing no significant difference in weight gain or loss between the two treatment groups ($n = 10$ each for tap and IEPA). d T2 weighted coronal image of one mouse which had been drinking IEPA for 2 days. The green square indicates the voxel in the bladder from which the spectrum of IEPA was obtained. e A reference spectrum of 200 mM IEPA obtained at 9.4 T. f Spectrum of IEPA obtained from the bladder (d) showing the same peaks as the reference spectrum, and indicating that the buffered IEPA was absorbed into the blood stream and filtered through the kidneys (rather than being directly excreted through the intestines)
as a bolus and can then be infused intravenously to establish high enough concentrations of IEPA at the site of pH measurement. Given as a 0.15 ml intraperitoneal bolus injection at 310 mM (as in [17]) this should result in a concentration detectable by MRSI but low enough to have no effect on the pH. However higher steady-state concentrations of IEPA can carry the risk that IEPA itself may alter the pH that is sought to be measured [19]. Additionally, to measure the pH, IEPA would also have to be infused in the non-treated animals. In order to ensure a distinct separation between treated and untreated animals, IEPA was used only as a buffer and not as an MRSI agent for the purposes of this study.

However, to assess whether ingested IEPA was absorbed into the systemic circulation, MR spectra were obtained from voxels within the bladder of animals that had been drinking IEPA for 2 days. These data showed that there was sufficient IEPA in the bladder (Fig. 1d) to obtain spectra (reference spectrum Fig. 1e, spectrum in bladder Fig. 1f), indicating that it was filtered through the kidneys, and thus was absorbed into the blood stream and systematically available.

Effect of IEPA on PC3M primary tumor growth, pH and spontaneous metastases

To investigate the effect of IEPA on primary tumor growth, we injected Luciferase expressing prostate cancer cells, PC3M-luc, subcutaneously into the right flank of male SCID mice. Four days prior to injections, half of the animals were provided with 200 mM IEPA in their drinking water. Animals were imaged weekly via bioluminescence. Images from representative animals (Fig. 2a) illustrate growth of the primary tumor in both cohorts of animals. However, the plot of bioluminescence of the primary tumor versus time (Fig. 2b) suggested a trend towards slower tumor growth in the IEPA treated animals. An unpaired two-tailed Students t-test, however, showed that these differences did not reach statistical significance ($P > 0.06$).

The tumor pH was taken immediately prior to euthanasia (29 days after injection of the primary tumor), and the tumors were harvested immediately afterwards. The mean tumor pH for each animal was averaged within each treatment group (tap versus IEPA); and the error was calculated as the standard error between the mean tumor pH in each group. Tumor pH measurements showed that, although the effects of IEPA were modest, the pH values of treated tumors were slightly higher ($P < 0.01$) than that of untreated tumors (Fig. 3a). Acidosis can induce both necrosis and apoptosis [20, 21]. Hence, in this system, we investigated whether IEPA treated tumors exhibited altered levels of cell death. Necrosis of the primary tumor were studied by H&E staining and counted by area, as outlined in Methods. The results of this analysis are shown in Table 1. Sample histological images are shown in Fig. 3b (Tap) and c (IEPA). These data indicate that the IEPA-treated tumors exhibited significantly less necrosis than controls (Fig. 3d; $P < 0.05$).

Metastasis is a process that includes a series of distinct steps starting with penetration of cancer cells through the basement membrane; local invasion into the surrounding tissue; intravasation to lymphatic or blood then extravasation and colonization in a new host organ [22]. To study the effect of IEPA on the entire metastasis paradigm, bioluminescent images were obtained from the thoracic regions of each cohort (IEPA vs. tap) of animals beginning 3 weeks after subcutaneous injection to form the primary tumor. These images were obtained in addition to the whole-body bioluminescence imaging which was used to monitor primary tumor growth. Because the primary tumors were much larger than the metastases, and therefore
had larger signals, the primary tumors were covered with opaque material during the lung imaging sessions. The thoracic region was imaged because the lungs are known to be one of the primary sites of metastases for this tumor model. Through these in vivo measurements, the thoracic region in the IEPA cohort was observed to have significantly fewer metastases compared to controls at 28 days (Fig. 4; \( P \leq 0.02 \)). Thus, although IEPA did not have a significant effect on primary tumor growth, it did reduce the numbers of spontaneous metastases in this system.

IEPA reduces experimental metastasis

The above data indicate that IEPA reduces spontaneous metastases, which involves both intravasation and extravasation, with only a moderate, non-significant effect on the primary tumor. This is consistent with previous studies using bicarbonate [13]. Because the effects of these buffers were limited mainly to the metastases, we then investigated whether IEPA affected the efficiency of extravasation and colonization with an experimental metastases model, wherein PC3M cells were injected directly into the tail vein of mice consuming either 200 mM IEPA or tap water. The results (Fig. 5) showed that IEPA led to a significant decrease in the appearance of lung metastases for up to 6 weeks after injection. These findings indicate that IEPA significantly inhibited extravasation and or colonization in this experimental metastases model with PC3M cells. After 6 weeks there were approximately \( 1 \times 10^8 \) photons/sec from the control mice and only ca. \( 3 \times 10^6 \) photons/sec emanating from the thoracic area of mice treated with IEPA (\( P \leq 0.002 \)).

Discussion

Previous work has shown that a volatile buffer, i.e. bicarbonate, was effective in inhibiting spontaneous and experimental metastases [13]. The aim of the current study...
was to investigate the effects of non-volatile buffers, specifically IEPA, to test the hypothesis that metastasis inhibition was due to buffering and was not specific to bicarbonate. Metastasis is the leading cause of death from cancer [23] and can be promoted by low pH and acidity; which is a common feature of solid tumors. The extracellular pH (pHe) of solid tumor is acidic, e.g. pHe of 6.5–7.1, compared to normal tissues with a pHe of 7.3–7.4. Experimental observations have shown that acid-mediated invasion can occur via destruction of the extracellular matrix, which is promoted by proteases and glycosidases and may convert in situ to locally invasive cancers [29]. An ‘acid-mediated invasion hypothesis’ was developed using mathematical models supported by empirical observations [14], and states that: (1) tumors export H\(^+\) into the adjacent stroma; (2) this results in damage to supporting stroma accompanied by cell death and extracellular matrix (ECM) break down; which (3) facilitates the invasion of the tumor cells and consequently colonization at distant sites. As a test of the model we observed that chronic oral administration of sodium bicarbonate (NaHCO\(_3\)), a physiologic buffer, was able to increase tumor pHe and reduce lung metastasis in spontaneous and experimental breast and prostate animal models. Neutralization of tumor acidity was measured with fluorescence microscopy of a dorsal window chamber and \(^{31}\)P magnetic resonance spectroscopy. Even though NaHCO\(_3\) was effective and inexpensive, its pK\(_a\) may be lower than optimal, and it was unclear whether it was working as a buffer, or if the effect was specific for bicarbonate. Therefore, in this study we examined a non-volatile buffer with a higher pK\(_a\) value for its effects on spontaneous metastasis.

We used IEPA with pK\(_a\) ~ 7, and the PC3M prostate cancer cell line was chosen because, as shown in our previous work, it was responsive to bicarbonate. IEPA was shown to be absorbed into the blood stream, and had no negative effects on the animal’s growth or behavior. In a subcutaneous tumor model, we found that IEPA only
sizable decrease in the growth rate of the primary tumor. However, this decrease in size was accompanied by a small but significant increase in the primary tumor pH and a substantial change in tumor anatomy; with a lower percentage necrosis when compared to control animals.

Although bone marrow is the most common site for prostate cancer metastasis, it is also known to metastasize to the lungs [24]. Spontaneous metastases, as measured by in vivo upper thoracic bioluminescence, were decreased in the IEPA-treated animals. Experimental metastases following tail vein injection were measured by total and thoracic in vivo bioluminescence, and these were also substantially decreased in the IEPA-treated animals. Hence, this non-volatile buffer was effective in reducing spontaneous and experimental metastases, compared to bicarbonate. This suggests that the effect is due to pH buffering and is not due to a specific bicarbonate effect per se. The fact that both spontaneous and experimental metastases were inhibited also suggests that the effect of the buffer therapy is on extravasation or/and colonization and not intravasation.

To investigate if the increase in tumor pH could be responsible for the remarkable decrease in metastases, we used the subcutaneous (spontaneous) model to measure the pH of the primary tumor. We observed a slight (statistically significant) increase in the primary tumor pH with IEPA treatment, accompanied by a substantial decrease in necrotic tumor anatomy. Although statistically significant, the change in pH of the primary tumor was very slight (0.06 pH units) and the physiological relevance of this change is unclear. In work with other tumor models, we have observed that the effect of bicarbonate on tumor pH was reduced in larger and/or rapidly metabolizing tumors (data not shown). The slight effects seen in this system are also consistent with reaction-diffusion models, which also predict that a larger and more substantial effect may take place at the site of colonization, when tumors are small (ca. 1 mm) and poorly perfused [14]. Unfortunately, no methods are available that can measure the pH of these small nascent colonies. Thus, we propose that the major effect of buffering occurs at metastatic sites, where acid load is lower due to smaller foci and where reducing acidity can inhibit local invasion [25–27].

Several mechanisms could potentially contribute to the effect of buffers on tumor metastasis. Angiogenic factors, such as vascular endothelial growth factor (VEGF-A) and interleukin 8 (IL-8) are upregulated by acidic pH [28]. Neovascularization influences the dissemination of cancer cells throughout the body ultimately leading to metastasis formation. IEPA may possibly function through increasing pH thereby affecting angiogenesis and consequently metastasis. The acid-mediated invasion can occur via destruction of the extracellular matrix, which is promoted by proteases and glycosidases and may convert in situ to locally invasive cancers [29]. Matrix metalloproteinases (MMP-2 and MMP-9) are believed to be critical for invasion. Even tough the MMPs have alkaline pH optima, low pH will increase their activation and release [11, 30]. Lysosomal proteases (cathepsins) are more active at acid pH, are released in invasive cancers, and function to proteolytically activate MMPs [31]. Thus, it is possible that the buffers, by neutralizing tumor acidity, are decreasing the proteolytic enzyme activation and prevent the ECM degradation which leads to decrease metastasis.

In summary significant reduction of metastasis with IEPA buffer support the hypothesis that non-volatile buffers with $pK_a \geq 7$ should be more effective in buffering extra cellular acidity [14]. This study specifically used prostate cancer model PC3M because in our previous work we have shown that this model was highly responsive to bicarbonate treatment. Previous work has shown that bicarbonate was ineffective in inhibiting metastases from B16 melanoma. These cells are more metabolically active and produce $H^+$ at a higher rate, compared to PC3M (data not shown) and hence, we hypothesize that they may be inhabitable with more buffering power, such as that of non-volatile buffers.

These findings suggest that non-volatile buffers, such as imidazoles may have the potential for clinical benefit. Sodium bicarbonate is currently in two clinical trials as a buffer; however compliance is low because of the taste. Thus, there is a need to find a palatable and effective alternative. The current work shows evidence that other buffers may be considered as an alternative to the NaHCO₃ in clinical trials with the possibility of better tolerance in addition to greater efficacy. This study indicated no toxicity in mouse models, in agreement with previous findings [32, 33]. Additionally, the measurements performed in the study by Garcia Martin et al. [15] indicate that the IEPA was able to extravasate only in the areas of the brain affected by the glioma and not in normal brain. Thus IEPA is not expected to cross the normal blood brain barrier within detectable limits.

In conclusion, this study demonstrates that a non-volatile buffer can reduce the incidence and growth of experimental metastases in a model of prostate cancer. Previous work using buffer therapy has exclusively used bicarbonate, which may act through alternative pathways, e.g. through activation of carbonic anhydrases. The current work demonstrates that alternative, non-volatile buffers can be equally efficacious as bicarbonate. Thus the mechanism of metastasis inhibition appears to be limited to pH buffering. It also introduces alternative buffers that may be more applicable in a clinical setting for acute or chronic use.
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