Assessment of Groundwater Quality and Saline Intrusions in Coastal Aquifers of Lagos Metropolis, Nigeria

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Abstract

As a result of immense industrialisation and high population growth, groundwater is heavily relied on in Lagos metropolis to serve as an alternative source of water where surface water is seriously polluted. The continued reliance on ground water has resulted in its decline in quantity and quality. In this study, the coastal aquifers of Lagos metropolis were selected for an assessment of its groundwater quality and impact of saline intrusion. Water samples collected along the coastal region were subjected to various physicochemical analyses. Results obtained were compared with permissible values for drinking water stated by Federal Environmental Protection Agency (FEPA) and World Health Organization (WHO). The results revealed that all the water samples were significantly hard (range 522.14–1233.34 mg/L). The salinity was delineated by conductivity measurements. Three samples had specific conductance above the stated limits for fresh water. The samples however met the stipulated limits for drinking water for the other tested parameters.

Keywords: Industrialisation, Groundwater, Physiochemical Analysis, Salinity, Conductivity

1. Introduction

Groundwater has long served as a source of drinking water and it is still very important today. The development of ground water has provided great socio-economic benefits to humanity. Globally, groundwater is estimated to provide about 50% of current drinking water supplies. As groundwater is isolated from the surface, most people take it for granted that groundwater should be relatively pure and free from pollutants. Although most groundwater are still of high quality, at some locations, it is becoming increasingly difficult to maintain the purity of groundwater. One of the major sources of pollution of groundwater is by saltwater intrusions. Others include seepages from underground storage tanks, oil wells, septic tanks, land fills and agricultural leaching.

Saline intrusions into coastal groundwater via aquifer penetration have become a major concern [1] because it is the commonest source of pollution to groundwater. The extent of saltwater intrusions into groundwater is dependent on several natural and anthropogenic factors; the nature of the aquifer and its natural recharge rates being the major natural factors. The anthropogenic factors include excessive groundwater withdrawals [1-6] and lack of sealing of abandoned boreholes and oil wells. Studies have indicated that increased groundwater salinity may be due to clearing of natural forests resulting in enhanced recharge which leaches salt downwards from salt stored in the unsaturated zone [7] or causes the water to rise dissolving salt as it does so [8]. The pollution of shallow aquifers under cities represents a major threat to sustainability of drinking water supplies in many urban areas throughout the world [4,9]. Frohlich and Urish [10] reported that the deterioration of the freshwater quality due to natural sea water infiltration affects the balanced life of the coastal strip of Rhodes Island. The present study was conducted in Lagos State in South western Nigeria, boarded in the south by the Atlantic ocean, in the north and east by Ogun state and in the west by Republic of Benin. It occupies an area of about 3.577 sq km with a population of about 14 million. 80% of the population resides in the metropolitan Lagos, making the state the most urbanized in Nigeria [11].

The aquifer structure of Lagos state falls in the Benin basin where salt water intrusion into the recent sediments aquifers occurs beneath the fresh water lens [12].
intrusions of salt water from the Atlantic ocean have caused untold hardship to habitats of the coastal areas of the state. There is an increasing pressure on the groundwater reserves of the state as a result of massive influx of people from other parts of Nigeria to Lagos metropolis (rural to urban migration). Longe et al. [13], estimate that about 10 million gallons of water are extracted from the multilayer aquifers areas per day. Most of the bore holes drilled in the areas of Lagos have been abandoned due to salt water intrusions into the aquifer.

The federal and state water agencies have initiated several hydrogeological evaluation of its groundwater resources [14,15]. However, the extent of groundwater intrusions and nature of these intrusions still require further investigations [16].

The objective of the present study is to ascertain the groundwater quality along the coastal region of Lagos state and assess the extent of saline intrusions in coastal aquifers of Lagos metropolis. Sampling sites were identified along the coastal line and groundwater samples were collected and subjected to physicochemical analysis.

2. Materials and Methods

2.1. Sampling

Groundwater samples were collected from eight (8) different sites (wells) along the coastal areas of Lagos metropolis. The wells were pumped for ten (10) minutes before samples were collected to ensure representative sampling. Plastic containers used for collection of samples were pre-treated by washing with 0.05 M HCl and then rinsed with distilled water [17]. After collection, sample bottles were tightly covered and transported under 4°C to the laboratory for chemical analysis. The location of sampling sites is shown in Figure 1. The global positioning coordinates as well as relief features of the sites are shown in Table 1.

Refrigerated samples were allowed to attain room temperature prior to chemical analysis. Fast changing parameters such as pH, temperature and electric conductivity were determined in situ. Temperature was measured using a mercury thermometer (range 0°C to 100°C), pH was measured using a pre-calibrated pH meter. Electrical...
Table 1. Showing sampling points, GPS coordinates and relative positions from the ocean.

| Location            | Code | Elevation (ft) | Dist. from ocean (m) | Latitude (°North) | Longitude (°) |
|---------------------|------|----------------|----------------------|------------------|---------------|
| Adeniji Adele       | AA   | 52             | 3600                 | 6.4572           | 3.3961        |
| Ajah                | AJ   | 57             | 2470                 | 6.4686           | 3.5924        |
| CMS                 | CMS  | 68             | 2180                 | 6.4541           | 3.3902        |
| Ebute Metta         | EBM  | 77             | 9000                 | 6.4903           | 3.3813        |
| Ikeja               | IKJ  | 91             | 4730                 | 6.6065           | 3.3482        |
| Ikorodu             | IKO  | 102            | 5000                 | 6.6411           | 3.4806        |
| Obalende            | OBA  | 116            | 5360                 | 6.4489           | 3.1482        |
| Victoria Island     | VIL  | 23             | 400                  | 6.4211           | 3.4806        |

Conductivity (specific conductance) was measured using a Horiba U90 meter with an accuracy of 0.001 µS/cm. Total dissolved solids, total hardness, sulphate, alkalinity and chloride were determined by standard methods [17, 18]. Alkali metals – sodium and potassium was determined by flame absorption spectrophotometry. All reagents used for analysis were prepared from analar grade chemicals. Appropriate reagent blanks were prepared for each analysis. All analysis was done in triplicates.

3. Results and Discussion

The results obtained from the analyses of ground water samples collected revealed that some of the tested parameters fall within the permissible limits stipulated by FEPA (Federal Environmental Protection Agency) [19] and WHO (World Health Organization) [20]. The pH, sulphate SO₄²⁻, phosphate PO₄³⁻, hardness due to calcium and alkalinity fall within the accepted values for drinking water as shown by the results in Table 2 and standards for FEPA and WHO in Table 3 and Table 4 respectively.

The total hardness of all the water samples however exceeds the permissible limits stated by the controlling agencies (refer to Tables 2, 3 and 4). The total hardness is a measure of contributions from calcium, magnesium, and other polyvalent cations such as iron, zinc, manganese, aluminium, and strontium. The low contribution of calcium to the total hardness may be suggestive of an absence of calcium bearing mineral from the underlying aquifer. The high values of total hardness may be due to the introduction of polyvalent cations into the groundwater system. All the water samples may be regarded as ‘very hard’ (hardness > 300 mg/L) according to the classification for hardness [18].

The chloride concentration measured for the samples did not exceed the limit set by WHO and FEPA (250 mg/L). Chloride is not considered harmful to humans, but imparts a salty taste to water above this limit. The high chloride concentration recorded in this study may however cause harm to plants if used for irrigation purposes. Chloride concentrations as high as 70 mg/L have been reported to cause damage to plant tissues [18]. The concentration of alkali metals sodium and potassium varied widely. Sodium concentrations ranged from 46.78 mg/L for Ebute Metta to 490 mg/L for Victoria Island. Potassium concentrations ranged from 8.67 mg/L for Ebute Metta to 341.70 for Victoria Island. The correlation R² (0.759), for both ions though weak is suggestive of identical sources for both ions.

Conductivity measurement is widely used as substitute for total dissolved solids (TDS) and salinity determinations. Adekunle [19] used resistivity method (inverse of conductivity) to delineate salt water intrusions into fresh water aquifer of Lekki Peninsula, Lagos, Nigeria. The data obtained for conductivity measurements in this study ranged from 126 µS/cm for Ikeja to 1565 µS/cm for Victoria Island. Permissible values for conductivity for drinking water is 1000 µS/cm. Values exceeding this limit are indicative of saline intrusions into the groundwater. Water samples collected from Adeniji Adele (1251.2 µS/cm), CMS (1520.0 µS/cm) and Victoria Island (1565.2 µS/cm) exceed the permissible value and could be under threat of saline pollution. The intrusions gradual intrusion of saline water into the groundwater for Adeniji Adele, CMS and Victoria Island may be attributed to their relative closeness to the Atlantic Ocean and their elevation from sea level. Data on relative elevation and distance from sea level is shown in Table 1. The groundwater in Victoria Island is under stress due to pollution from saline waters. Characteristic features of the area include: close proximity to the ocean, low elevation from sea level, poor aquifer structure (belonging to the Benin basin where salt water bearing sand overlie fresh water [12]. Adekunle [19]; Oteri and Atolagbe [12] have previously reported cases of saline intrusions occurring in Lekki Peninsula which is separated from Victoria Island by Falomo River. Obalende which has the highest eleva-
Table 2. Showing mean results of physico-chemical analyses of samples.

| Code | pH  | (SO₄)²⁻ mg/L | (PO₄)³⁻ mg/L | Ca²⁺ mg/L | Total Hardness mg/L | Cl⁻ mg/L | Na⁺ mg/L | K⁺ mg/L | Electric. Cond. µs/cm | TDS mg/L |
|------|-----|--------------|--------------|-----------|--------------------|---------|---------|---------|----------------------|---------|
| AA   | 6.01| 34.91        | 0.01         | 120.9     | 1090.25            | 94.3    | 232.82  | 86.67   | 1251                 | 625.50  |
| AJ   | 6.38| 29.68        | ND           | 89.24     | 998.56            | 61.28   | 156.99  | 36.01   | 407                  | 203.51  |
| CMS  | 6.68| 63.4         | ND           | 121.2     | 1233.34           | 58.2    | 483.42  | 193.2   | 1520                 | 760.01  |
| EBM  | 6.76| 24.28        | 0.03         | 27.98     | 522.14            | 22.33   | 46.78   | 8.67    | 130                  | 65.05   |
| IKJ  | 6.59| 32.62        | ND           | 41.73     | 526.14            | 19.01   | 56.15   | 10.91   | 126                  | 63.31   |
| IKO  | 6.45| 32.71        | ND           | 47.32     | 890.53            | 28.98   | 91.99   | 20.46   | 288                  | 144.13  |
| OBA  | 5.93| 44.07        | ND           | 148.76    | 1131.46           | 92.16   | 311.24  | 92.88   | 940                  | 470.00  |
| VIL  | 6.63| 42.13        | ND           | 119.2     | 926.25            | 109.5   | 430.07  | 341.7   | 1565                 | 782.51  |

ND = Not detected

Table 3. Federal Environmental Protection Agency (FEPA) now Federal Ministry of Environment - water quality criteria (FEPA, 1988).

| Parameter | Permissible criteria (mg/L) | Desirable criteria mg/L |
|-----------|-----------------------------|-------------------------|
| Colour    | 75                          | < 10                    |
| Odour     | Virtually absent            | Virtually absent        |
| Turbidity | 25 Turbidity unity          | Nil                     |
| pH        | 6.0-8.5                     | 6.0-8.5                 |
| Alkalinity CaCO₃ | 30-500             | 30-500                 |
| Cl⁻       | 2.5                         | < 2.5                   |
| SO₄²⁻     | 250                         | < 50                    |
| TDS       | 500                         | 500                     |
| P         | 0.01                        | 0.01                    |

Table 4. World Health Organisation (WHO) standards for drinking water (WHO, 1971).

| Parameter | Maximum permissible limit |
|-----------|---------------------------|
| Turbidity | 25 turbidity units        |
| pH        | 6.5-9.2                   |
| Total solids | 1500 mg/L            |
| Total Hardness | 500 mg/L                |
| Hardness-Ca | 200 mg/L              |
| Cl⁻       | 600 mg/L                  |
| SO₄²⁻     | 400 mg/L                  |

Table 5. Assessment of water samples based on the stipulated limits.

| Location | pH | SO₄²⁻ | PO₄³⁻ | Hardness-Ca | Cl⁻ | Total Hardness | Electric. Conductivity |
|----------|----|-------|-------|-------------|-----|----------------|------------------------|
| AA       | +  | +     | +     | +           | +   | −              | −                      |
| AJ       | +  | +     | +     | +           | +   | −              | +                      |
| CMS      | +  | +     | +     | +           | +   | −              | −                      |
| EBM      | +  | +     | +     | +           | +   | −              | +                      |
| IKJ      | +  | +     | +     | +           | +   | −              | −                      |
| IKO      | +  | +     | +     | +           | +   | −              | +                      |
| OBA      | +  | +     | +     | +           | +   | −              | +                      |
| VIL      | +  | +     | +     | +           | +   | −              | −                      |

+ meets the stipulated permissible limits
− fails to meet the stipulated permissible limits
tion from sea level (Table 5) still has high electrical conductivity. This may be attributed to the area’s close proximity to the ocean (3.6 Km) as compared to values for the other sites.

4. Conclusions

Conclusively, at present there is no universal system for classification of groundwater systems. Several agencies such as the Rhode Island department of environmental management office of water resource have classified groundwater based on its suitability for drinking with or without pre-treatment. The assessment of groundwater for Lagos state adopted in this study is based on whether the water meets all the requirements as stated by FEPA and WHO. The result from this study revealed that all the water samples analysed were hard and three did not meet the standard for conductivity. The water samples however met the permissible limits for drinking water for SO$_4^{2-}$, PO$_4^{3-}$, Cl$^-$, pH and hardness due to calcium.

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