The Experimental Study of The Effect of Air Preheating in MSW Pellet Multi-Stage Downdraft Gasifier

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Abstract—High municipal solid waste generation in Indonesia can be used to substitute fossil energy. MSW is converted into pellet form to uniform moisture content. This study aimed to improve the performance of gasified downdraft gasifier processes by heating air inlets in pyrolysis, oxidation, and reduction zones. Variations of air temperature inlet in these zones are 80 °C, 110 °C, 150 °C, and 210 °C. The results of this study obtained the highest air temperature at 210 °C for 969 °C. The syn-gas compositions (CO and H₂) at an increase in temperature of 80-210 °C increased from 21.4%, 9.99%, and 1.77% to 22.79%, 10.28%, and 1.79% respectively. Cold gas efficiency increased by 5.13 % and the lowest tar level was 34.39 mg/Nm³. Based on the result of research, preheated air can improve the efficiency of gasification and reduce tar content in the syngas.

Keywords—MSW pellet, gasification, preheated air, multi-stage

I. INTRODUCTION

MSW (Municipal Solid Waste) is an urban solid waste that we often find every day. At present, MSW utilisation is still very low. One way to convert MSW into energy is gasification technology. To uniform the moisture content in MSW, the MSW is converted into a pellet first. Gasification is a thermochemical process that converts biomass into useful gas by a partial oxidation process using air, oxygen or steam [1]. In general, the gasification process is divided into four stages: (1) drying process (endothermic stage), (2) pyrolysis (endothermic stage), (3) partial oxidation (exothermic stage) and, (4) reduction process (endothermic stage). The gasification product is called synthetic gas (syngas) containing CO, H₂, and CH₄ as a combustible gas. In this research, the gasification process using downdraft type gasifier. Downdraft gasifier has several advantages, namely high carbon conversion rate, low tar production, and simple reactor design [2].

Several studies have been conducted to improve the efficiency of gasification and reduce tar levels in the syngas. One of them is by heating the air used as an internal gasification agent (heated in the gasifier) and externally (heated before entering the gasifier). Bhattacharya and Dutta [3] preheat air at an optimum temperature of 210 °C, the resulting oxidation zone temperature of 1000 °C, and at this temperature, the tar level in syngas can be reduced to 10 mg/Nm³. In addition, according to a study by Guanggul [4], heating of air prior to entering the oxidation zone to 200 °C temperature can increase the quality of syngas produced by increasing volumetric percentage H₂ from 8.47% to 10.53%, CO from 22.87% to 24.94%, CH₄ from 2.02% to 2.03%, and HHV from 4.66 MJ/Nm³ to 5.31 MJ/Nm³. Sudarmanta performs gasification characterisation, the results of his research show the effect of heating the inlet air of the oxidation zone with the external heating to the temperature distribution at the reactor, in particular, the oxidation zone [5]. The study showed that air heating from 30 °C to 70 °C increased the temperature of the oxidation zone from 710 °C to 800 °C, which by increasing the temperature reduces tar levels and increases the quality of syngas (increases the syngas energy from 41,311 kJ to 44,001 kJ). Then on downdraft reactor which is currently in the laboratory of Combustion and Energy Systems has been conducted by Akbar [6] concerning temperature distribution at reactor zones with the existence of internal air heater in the form of air pipe belt wrapped around the oxidation zone. From the research obtained temperature in the oxidation zone of 800 °C and LHV of 3.699 kJ/kg. Based on these explanations, the addition of an external air heater (preheated water) can increase the desired gasification temperature. However, a study conducted by Sudarmanta [5] and Akbar [6] only heated the air up to 70°C while the optimum temperature of warming reached 210 °C. Therefore, this study will add an external air heater from 80 °C up to 210 °C before the air enters the internal heating to be supplied into the downdraft reactor oxidation zone with MSW pellet feedstock. So it is expected to increase the composition of combustible syngas (CO, H₂, and CH₄) and reduce tar content.

II. METHOD

The raw materials used in this research are MSW pellets 15 cm in length and 8 mm in diameter. MSW pellet composition comprises 60% organic matter and 40% inorganic [7]. Then, in this study, downdraft type reactors with multilevel air inputs in pyrolysis, oxidation and reduction zones [8][9], and [10] were modified by adding external heaters to the inlet air. This external heater comes from an open coil heater that is installed in the air input pipe. The addition of external heaters is done because internal heaters in the pyrolysis, oxidation, and reduction zones are not able to increase air temperature to 210 °C. Experiments are designed with schemes like Figure. 1.

Experiments were conducted with five variations of air temperature in pyrolysis, oxidation and reduction zones, 80 °C, 110 °C, 150 °C, 180 °C, and 210 °C. Air ratio and equivalence ratio are kept constant at 1:8:1 and 0.4 respectively. Measurement of temperature distribution using thermocouples mounted along the reactor (T1 to
T7) in Figure. 1. Then the composition of combustible syngas was measured using gas chromatography, and samples were taken at syngas output (shown with no 15 in Figure. 1). The content of tar in the syngas is measured using a tar condenser.

III. RESULTS AND DISCUSSION

The MSW pellets have characteristics as shown in Table. 1. These characteristics are used to calculate the equivalence ratio and the input energy of the gasification process. Next, the performance of a gasifier in generating combustible gas can be observed from several parameters, including temperature distribution throughout the reactor (drying zone, pyrolysis zone, oxidation zone and reduction zone), combustible syngas composition, cold gas efficiency, and tar content. These parameters were reviewed with variations of multi-stage inlet air temperature and analysed its effect.

A. Temperature Distribution along Gasifier

Figure 2 shows thermocouples (T1) and (T2) at a temperature interval of 50 to 150 °C. The interval shows that at an altitude of 90 - 120 cm there has been evaporation of water content. Then, T1 and T2 are located in the drying zone, where all the water content in MSW has run out. Also, the temperature at the oxidation zone (T6) has increased significantly along with the increase in air temperature. This is due to the high air temperature makes the reaction of partial oxidation, and char combustion in the oxidation zone becomes more reactive. Also, heating the air can increase the enthalpy of oxygen as a reactant, thus making the energy difference of products and reactants grow larger. The
highest oxidation temperature was obtained at the variation of air temperature 210 °C equal to 969 °C. Temperature increases also occur in T5 and T7 which are pyrolysis zone temperatures and reduction zones. Two factors cause temperature increases in pyrolysis zones and reduction zones. First, the heating of the air makes the oxidative reaction of pyrolysis, and the heterogeneous oxidation reaction in the reduction zone becomes more reactive. Second, an increase in the temperature of the oxidation zone (exothermic) causes heat transfer to the pyrolysis zone and the higher the reduction. Highest pyrolysis and reduction temperatures were obtained at variations of air temperature 210 °C for 569 °C and 644 °C respectively.

The gasifier temperature increases when the air temperature (gasiﬁer agent) is heated. The highest temperature increase occurs at the T6 thermocouple point located in the oxidation zone. The increase in temperature on T6 when the air temperature is increased to 210 °C, is 55 °C. Besides that, the increase in gasifier temperature also occurs at the thermocouple points T3, T4, T5, and T7. When the air temperature is heated to 210 °C, the increase in temperature at the thermocouple points T3, T4, T5, and T7 is 19, 26, 45, and 49 °C respectively. Whereas at the thermocouple point T1 and T2 the gasiﬁer temperature increase due to air heating is not signiﬁcant. This is because the thermocouple points T1 and T2 are located far from the heat source (oxidation zone) and also there is no air entering this zone.

The increase in gasiﬁer temperature is caused by an increase in enthalpies of air as a reactant which causes changes in the enthalpies formation to become large. That makes the reaction speed increase following the equation in Figure 2.7. Then, in this study the ratio of air into the gasiﬁer is 1: 8: 1, which means that 10% of the air enters the pyrolysis zone, 80% to the oxidation zone, and 10% to the reduction zone. This comparison causes a difference in the increase in gasiﬁer temperature. The high percentage of air intake results in a very signiﬁcant effect of heating the air in the oxidation zone.

The effects of rising temperatures on pyrolysis, oxidation, and reduction zones are the increasing of combustible gas compositions, especially CO and H2. Then, increasing the temperature of these zones can also reduce the tar content in the syngas.

### B. Syngas Composition

Figure 3 represents the volumetric percentage value of the CO, H2, CH4, and O2 compounds present in the syngas at each variation of air temperature. At 59-210 °C increase in air temperature, the volumetric percentage of CO and H2 compounds increased by 1.7% and 0.31%. Increased volumetric CO percentage is caused by increasing rate of oxidation reaction in three gasification zones (pyrolysis, oxidation, and reduction). The relation between the reaction rate and the reaction temperature is shown in Equations (1) [1].

\[ k = A_0 \exp \left( - \frac{E}{RT} \right) \]  \hspace{1cm} (1)

k is the constant of the reaction rate, \( A_0 \) is the pre exponential constant, R is the universal gas constant, E is the activation energy, and T is the reaction temperature.

Then the volumetric percentage of CH4 did not increase signiﬁcantly when it was raised at 59-200 °C. This is because the reaction rate of methanation in the reduction zone \(( \text{C} + 2\text{H}_2 \rightarrow \text{CH}_4 \)) occurs very slowly [3]. So the inlet air temperature is not so inﬂuential on the formation of CH4.

Increasing the CO and H2 composition in Figure 3 along with increasing air temperature, followed by decreasing O2 composition which is non-ﬂammable syngas. The O2 composition of syngas is oxygen derived from the inlet air of an unreacted oxidation zone. Therefore, the decrease in the O2 composition is due to increased air temperature inlet to the gasiﬁcation zone. Then, according to Basu [3], In a char combustion reaction in the oxidation zone there is a partition coefﬁcient (\( \beta \)) which has a relationship with the surface temperature of the charcoal (Toxidation) with Equation (2) [11] as follows:

\[ \beta = \frac{[\text{CO}]}{[\text{CO}_2]} = 24000e^{-\frac{6238}{T}} \]  \hspace{1cm} (2)

T is the surface temperature of the charcoal (K), which in this study is a measured temperature at T6.

Increased compositions of combustible syngas (CO and \( \text{H}_2 \)) make the syncopated lower calorific value (LHV) syngas increased. This indicates that warming of air temperature can increase the energy of gasification process output. Then also the increased energy output of MSW pellet gasiﬁcation makes the gasification process more efﬁcient.

### C. Cold Gas Efﬁciency

The performance parameter of the next gasiﬁcation process is cold gas efﬁciency. Figure. 4(a) shows that with an increase in air temperature of 59-210 °C it can increase the cold gas efﬁciency by 6.07% with a relatively linear increase. The increase is due to the increase of LHV in the syngas. The volumic rate of

### Table 1. Proximate Characteristics of the Pellet MSW

| Composition (% wt) | Moisture content | Ash content | Volatile matter | Fixed carbon | Ultimate analysis (% wt) |
|--------------------|------------------|-------------|-----------------|--------------|--------------------------|
|                    | 10.23            | 5.73        | 70.93           | 13.11        | C: 50.34, H: 4.91, O: 38.83, N: 0.13, S: 0.06 |
| HHV                | 16.88 MJ/kg      |             |                 |              | LHV: 15.69 MJ/kg         |
| LHV                | 16.88 MJ/kg      |             |                 |              |                         |

**Figure 3. Composition and LHV of syngas**
Syngas also influences it. Can be seen in Figure. 4(a), when the air temperature is increased, the volumetric rate of syngas also increases. This indicates that with the increase in air temperature can maximise the reaction between biomass and air, which is characterised by an increase in the volumetric rate of syngas.

Figure. 4(a) also shows the trend value of cold gas efficiency on heating air temperature 80-210 °C not found peak value or curve of upward trend becomes a downward trend. This indicates, the value of cold gas efficiency at air temperature exceeding 200 °C still possible to increase to a specific temperature.

D. Tar Content

The use of allowed syngas for diesel engines should have a tar content below 100 mg/Nm³ [12][13], and [14]. Therefore, the output parameter of the gasification process that needs to be analysed is the tar content in the syngas. Figure. 4(b) shows a graph of the effect of increasing air temperature on tar content in the syngas.

In Figure. 4(b), the tar content decreased in 80-210 °C air temperature variations by 70%. The addition of a heater to increase the temperature of the inlet air of the oxidation zone affects the temperature of the gasifier thus causing the reduction of the tar content in syngas by a process called thermal cracking. Thermal cracking is a process used to reduce tar content by increasing temperature.

The content of tar in the syngas is strongly influenced by the temperature of the gasifier. The higher the temperature value of the gasifier, the tar content in syngas will be smaller. The gasifier temperature in this study is above 900 °C, so the type of tar that dominates is the tertiary and secondary tar [1]. Tertiary tars have lower molecular mass than secondary tars, and primary tars, the tar content present in syngas at higher gasifier temperatures will have a lower tar content.

IV. CONCLUSION

Based on the results obtained from the research conducted, it can be concluded as follows:

1. Increase inlet air temperature of pyrolysis, oxidation, and reduction zones, from 80 °C to 210 °C increases the temperature of the oxidation zone from 914 °C to 969 °C, the pyrolysis temperature from 524 to 569 °C, and the reduction zone temperature from 595 °C to 644 °C
2. The gasification process in MSW pellets with the addition of external heaters can increase the volumetric percentage of combustible syngas (CO and H₂) from 21.4% and 9.99% to 22.79%, and 10.28% respectively
3. Cold gas efficiency increases linearly with increasing temperature of the inlet air of the oxidation zone. Increased air temperature (80-210 °C) due to the addition of external heaters to make cold gas efficiency increased by 5.13%.
4. The tar content in the lowest syngas is worth 34.39 mg/Nm³, i.e. at 210 °C inlet air temperature.

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REFERENCES

[1] P. Basu, Biomass Gasification, Pyrolysis, and Torrefaction. India: Academic Press, 2013.
[2] A. Molino, S. Chianese, and D. Musmarrà, “Biomass Gasification Technology: The State of The Art Overview,” J. Energy Chem., vol. 25, pp. 10–25, 2016.
[3] Bhattacharya and A. Dutta, “Two-Stage Gasification Of Wood With Preheated Air Supply: A Promising Technique For Producing Gas Of Low Tar Content,” Bioresour. Technol., vol. 126, pp. 224–232, 2012.
[4] F. M. Guanggul, S. A. Sulaiman, and A. Ramli, “Gasifier Selection, Design and Gasification of Oil Palm Fronds with Preheated and Unheated Gasifying Air,” Bioresour. Technol., vol. 126, pp. 224–232, 2012.
[5] B. Sudarmanta, D. Muratji, and D. Walansari, “Karacterisasi Biomassa Sekam Padi Menggunakan Reaktor Downdraft dengan Dua Tingkat Lahan Udara,” Surabaya, 2009.
[6] F. Akbar and B. Sudarmanta, “Studi Eksperimental Pengaruh Air Fuel Ratio Proses Gasifikasi Briket Municipal Solid Waste Terhadap Unjuk Kerja Gasifier Type Downdraft,” Surabaya, 2016.
[7] Indarto, “Karacterisasi unjuk Kerja Mesin diesel generator set sistem dual fuel solar dan biogas dengan penambahan fan Udara sebagai penyuplai Udara,” Surabaya, 2016.
[8] Bhattacharya and et al., “A study on multi-stage hybrid gasifier engine system,” Biomass and Bioenergy, vol. 21, pp. 445–460, 2011.
[9] B. Sudarmanta, A. Gafur, A. Saleh, R. Dwiyantoro, and Sampurno., “The effect of two stage gasifying agent on
biomass downdraft gasification to the gasifier performance,” in *AIP Conference Proceedings*, 2018.

[10] A. Saleh and B. Sudarmanta, “Experimental investigation on multi-stage downdraft gasification: Influence of air ratio and equivalent ratio to the gasifier performance,” in *AIP Conference Proceedings*, 2018.

[11] J. Arthur, “Reaction between carbon and oxygen,” *Trans Faraday Soc*, vol. 47, pp. 164–178, 1951.

[12] M. Brown, G. Mudge, and B. K., “Evaluation of processes for removal of particulates, tars, and oil from biomass gasifier product gases,” in *Energy from Biomass and Wastes X*, 1987.

[13] A. Bridgewater, “The Technical and Economic Feasibility of Biomass Gasification for Power Generation,” *Fuel*, vol. 74, no. 5, pp. 631–653, 1995.

[14] H. Stassen, “UNDP/WB Small Scale Biomass Gasifier Monitoring Report,” Netherlands, 1993.