Formulation and Optimization of Gentamicin Hydrogel Infused with Tetracarpidium Conophorum Extract via a Central Composite Design for Topical Delivery

Gentamisin Hidrojelinin Tetrakarpidium Conophorum Ekstraktı ile Aşılanarak Formülasyonu ve Optimizasyonu

Objectives: Response surface methodology coupled with statistically designed experiments has been found to be very useful in optimising multivariable processes. The aim of this study was to evaluate the influence of two independent variables, a ratio of permeation enhancers/antioxidants (transcutol and ethanolic extract of tetracarpidium conophorum EETC) and stirring rate, on the flux and permeation of gentamicin hydrogel.

Materials and Methods: A modification of free radical initial polymerization was used to formulate the gentamicin hydrogel. A 32 factorial CCD was then used to investigate the effect of independent variables of the permeation enhancer transcutol: EETC (X1), stirring speed (X2) via 14 formulation batches, which were evaluated for dependent variables flux (Y1) and amount of drug permeated after 12 hours (Y2) ex vivo.

Results: The results of ANOVA performed to determine the fit of the models revealed that the models were statistically significant (p<0.05) and did not show lack of fit (R2>0.80). The regression equation generated for flux was Y1=19.35 – 25.82X1 – 0.044X2 + 0.0097X1X2 + 11.86X21 and for cumulative permeation of gentamicin in 12 hours Y2=315.50 – 189.67X1 + 0.28X2 -1.29X1X2 + 123.55X21. The validity of the statistical models used for predicting flux and drug permeation was confirmed by conducting three confirmation experimental runs at the identified optimum conditions. The results showed that there was no significant difference between the experimental results and those predicted by the statistical models.

Conclusion: The excellent correlation between the predicted and measured values shows the validity of statistical models (R2=0.95). An antioxidant and permeation enhancer has been used for the first time to investigate the influence on dependent variables. Optimization of gentamicin hydrogel using central composite statistical design is valid for the prediction of drug permeation and flux using variables in formulation.

Key words: Hydrogels, polymer, Tetracarpidium conophorum, response surface methodology, gentamicin

Amaç: İstatistiksel olarak tasarlanmış deneyler ile birleştirilmiş cevap yüzey metodolojisinin, çok değişkenli proseslerin optimize edilmesinde çok yararlı olduğu bulunmuştur. Bu çalışmamızın amacı, iki bağımsız değişkenin, permeasyon artırıcı/antioksidanların (transcutol ve tetrakarpidium conophorum EETC) ve karıştırma hızının, gentamisin hidrojelinin akışı ve permeasyonu üzerindeki etkisini değerlendirmektir.

Gereç ve Yöntemler: Gentamisin hidrojelinin formülé etmek için serbest radikal başlangıç polimerizasyonu kullanıldı. Daha sonra, permeasyon artırıcı transkutol: EETC (X1), karıştırma hızı (X2) bağımsız değişkenlerinin etkisini araştırmak için, bağımsız değişkenler, ağız (Y1) ve 12 saat ex vivo permeasyondan sonra elde edilen ilaç miktarı (Y2) için değerlendirilerek, 14 formülasyon grubu, 32 faktöriyel merkez CCD kullanılmıştır.
In studies by Ezealisiji et al. of alkaloids, saponins, glycosides, flavonoids, and tannins extracts of this plant, the utility of RSM via CCD for optimizing topical gentamicin hydrogel production was explored. In this investigation, we utilized RSM via CCD for optimizing topical gentamicin hydrogel production using a two variable CCDs via free radical initial polymerization of the alkyd acrylate polymer. The developed optimised formulation was evaluated for performance-related in vitro drug release and ex vivo permeation study. Physicochemical characterization of the gel was conducted via rheologic studies, drug content evaluation, Fourier-transform infrared spectroscopy (FTIR), and the mechanism of release was evaluated via varying kinetic models.

**INTRODUCTION**

Gentamicin is a water-soluble aminoglycoside antibiotic derived from *Micromonaspora purpurea*, and actinomycete. It is used for the treatment of infections caused by susceptible strains of *Pseudomonas aeruginosa*, *Proteus* (species: positive and indole-negative), *Escherichia coli*, *Klebsiella-Enterobacter-Serratia* species, *Citrobacter* species, and *Staphylococcus* species (coagulase-positive and coagulase-negative). When required for topical administration, it is usually formulated as creams because it calls for compatibility of the gel with human natural tissue without causing any toxicity upon its degradation.1 The water holding capacity and permeability are the most important characteristic features of a hydrogel.2 Biocompatibility is the third most important characteristic required by a hydrogel because it calls for compatibility of the gel with human natural tissue without causing any toxicity upon its degradation.2 In addition to the above characteristics, the soft and rubbery nature of hydrogels minimises irritation to surrounding tissue. Their highly porous structure, which can easily be tuned by controlling the density of the cross-links in the gel matrix and the affinity of the hydrogels for the aqueous environment in which they are swollen,3 is also an advantage. The porosity of hydrogels also permits loading of drugs into the gel matrix and subsequent drug release at a rate that is dependent on the diffusion coefficient of the small molecule or macromolecule through the gel network.3

*Tetracarpidium conophorum*, commonly called the African walnut plant, whose ethanolic extract would make up the second component of the formulation of study, is a perennial climbing shrub 10 to 20 feet high, found growing wild in forest zones of sub-Saharan Africa, including Nigeria. Studies have shown that the African walnut possess some beneficial antibacterial,4 antioxidant,5 and immune-stimulating properties. It is commonly used in Nigerian folkloric medicine for the treatment of bacterial infections and ailments caused by oxidative stress.6 Photochemical screening of ethanolic extracts of *Tetracarpidium conophorum* showed presence of alkaloids, saponins, glycosides, flavonoids, and tannins in studies by Ezealisiji et al. The antibacterial properties of this plant extract can be attributed to the presence of these secondary metabolites. By incorporating these extracts into a three-dimensional polymer network formed by hydrophilic polymer chains via either physical or chemical bonds, hydrogels will be used to form a novel drug delivery system comprising components that will work synergistically to facilitate wound healing.7

In the development of a topical dose form, an important issue was to design an optimized pharmaceutical formulation with an appropriate penetration rate within a short time period with minimum trials. Traditional experiments require more effort, time, and materials when a complex formulation needs to be developed. Recently, response surface methodology (RSM) via central composite design (CCD) coupled with statistically designed experiments has been found to be very useful in optimising multivariable processes and it has been successfully applied to the optimisation of many bioprocesses.8,9 Based on the principle of design of experiments, the methodology encompasses the use of various types of experimental designs, generation of polynomial equations, and mapping of the response over the experimental domain.

In this investigation, we explored the utility of RSM via CCD for the optimization of topical gentamicin hydrogel production using a two variable CCDs via free radical initial polymerization of the alkyd acrylate polymer. The developed optimised formulation was evaluated for performance-related in vitro drug release and ex vivo permeation study. Physicochemical characterization of the gel was conducted via rheologic studies, drug content evaluation, Fourier-transform infrared spectroscopy (FTIR), and the mechanism of release was evaluated via varying kinetic models.

**EXPERIMENTAL**

**Chemical and reagents**

Gentamicin sulphate (BP grade) was obtained as a gift from Drugfield Pharmaceuticals Limited (Ogun State, Nigeria), Carbopol Ultrez 21® was obtained as a gift from Metchem Limited (Mumbai India/Lubrizol Corporation, USA), Carbopol 940® (Lubrizol Corporation, USA), propylene glycol, triethanolamine (TEA) (Merck Germany), Transcutol® was obtained as a gift from Gattefosse (Cedex, France), O-Phthalaldehyde OPA from Fluka (Steheim Germany). N-acetyl cysteine (NaC) sodium hydroxide was from Sigma Aldrich (St. Louis, USA). All other chemical and reagents were of analytical grade.

**Extraction of ethanolic extract of Tetracarpidium conophorum (EETC)**

The plant was collected from farms in Nkwere Local Government Area, Imo state, Nigeria, and identified by Mr Oyebanji O.O of
were analysed using Design Expert for optimisation using RSM. The experimental observations
interactions. were observed to access any possible physical or chemical
and the appearance (or disappearance) of peaks in the spectra
analysed via FTIR. The physical appearance of the samples
South Africa). The optimized hydrogel formulation was also
recorded in the range of 400-4000/cm using the potassium
compressed to form pellets with a hydraulic press at 10 tons’
were separately mixed with three parts of potassium bromide and
the polymers (Carbopol Ultrez 21) and excipients (1:1) were
was evaluated using FTIR. Physical mixtures of gentamicin,
and the gel fraction were then mixed at a varying stirring rates
to form the polymeric hydrogel. The hydrogels were stored in
sealed glass containers for further analysis.

Preparation of gentamicin-loaded acrylate copolymer based hydrogels
Gentamicin sulphate (0.1% w/w) was dissolved in aliquots
of purified water and propylene glycol was titrated in drops
into the mixture. The permeation enhancer Transcutol:EE TC
(antioxidant extract) in varying ratios 2% v/v and 10% v/v
propylene glycol were incorporated into the aqueous phase
of the formulation. At 25°C, the gel phase was prepared by
dispersing the alkyl acrylate cross-polymers Carbopol® Ultrez
21 (1.5% w/v) in purified water using a mechanical stirrer at
a predetermined stirring rate. The pH was adjusted with the
cross-linking agent TEA to a pH of 5.5. Both the aqueous fraction
and the gel fraction were then mixed at a varying stirring rates
to form the polymeric hydrogel. The hydrogels were stored in

FTIR
Gentamicin, EETC, and Carbopol Ultez 21® compatibility
was evaluated using FTIR. Physical mixtures of gentamicin,
the polymers (Carbopol Ultrez 21) and excipients (1:1) were
separately mixed with three parts of potassium bromide and
compressed to form pellets with a hydraulic press at 10 tons’
presence. The FTIR absorption spectra of all samples were
recorded in the range of 400-4000/cm using the potassium
bromide disc method with FTIR spectroscopy (Bruker,
South Africa). The optimized hydrogel formulation was also
analysed via FTIR. The physical appearance of the samples
and the appearance (or disappearance) of peaks in the spectra
were observed to access any possible physical or chemical
interactions.

Experimental design
A two-variable, CCD was used for the formulation of the
gentamicin hydrogels. The independent variables tested
included the Transcutol:EE TC ratio and stirring speed. These
variables were varied over five levels and replicated six times
at the centre point to result in a total of fourteen experimental
runs. The ranges of the independent variables are shown in
Table 1. Two responses, namely flux (µg/cm²/hr) and the
amount of drug permeated after 12 hours (µg/cm²) were chosen
for optimisation using RSM. The experimental observations
were analysed using Design Expert® 7.0.0 software (Stat-ease,
Inc. Minneapolis, USA). The coded and actual values of the
independent variables were calculated using Equation (1).

\[ X_i = \frac{X_i - X_0}{\Delta X_i} \quad (1) \]

Where \( X_i \) and \( X_0 \) are the coded and actual values of the
independent variable, respectively. \( X_i \) is the actual value of the
independent variable at the centre point and \( \Delta X_i \) is the step
change in the actual value of the independent variable. The experimental data was fitted according to Equation (2) as a
second-order polynomial equation including the main effects
and interaction effects of each variable. One-way analysis of
variance (ANOVA) and response surface plots were generated
using Design Expert and the optimised value of the independent
variables for optimum response was determined using
numerical optimisation.

\[ Y_i = b_0 + \sum b_{ij} X_i + \sum b_{i} X_i X_j + e_i \quad (2) \]

where \( Y_i \) is the dependent variable or predicted response, \( X_i \)
and \( X_j \) are the independent variables, \( b_{ij} \) is offset term, \( b_i \) and \( b_j \)
are the single and interaction effect coefficients, and \( e_i \) is the
error term.

Physical evaluation of hydrogel formulation
The hydrogels were physically examined for colour,
homogeneity, and consistency.

pH evaluation
The pH of the hydrogels was recorded using a pH meter
(Ashford, UK), ensuring that the electrode was in contact with
the formulated hydrogel for 45 seconds to allow for equilibration.
Experiments were performed in triplicate.

Rheologic studies
The viscosities of the varying formulations were determined at
25°C at varying rpm with the aid of a cone and plate viscometer
with spindle-4, (Brookfield Engineering Laboratories, DV-E
Digital viscometer ID:12020N15).

Drug content determination
One gram of hydrogel was dissolved in 10mL of water, centrifuged
at 500 rpm for 45 mins, and filtered using a 0.5-µm millipore
filter. Using a 1:50 dilution, the concentration of gentamicin was
obtained using a ultraviolet (UV)/visible spectrophotometer
(UV-Vis 2600 Shimadzu Analytical and measuring instruments)
after derivatisation using O-Pthalaldehyde reagent with
Kowalcuz’s method.\(^{(10)}\) Phthalaldehyde reagent was formulated
prior to use by dissolving 20 mg of O-Pthalaldehyde in 1.0

Table 1. Experimental range of independent variables for a two
factor CCD

| Independent variables | Symbols | Coded and actual levels |
|-----------------------|---------|------------------------|
| Transcutol:EE TC (-)  | \( X_1 \) | -1.414 -1 0 1 1.41     |
| Stirring speed (rpm)  | \( X_2 \) | 1/3 3/7 2/3 0.902 1     |
| EETC: Ethanolic extract of Tetracarpidium conophorum | CCD: Central composite design |

1.41
mL of methanol to 1.5 mL of a 10% NaC and diluting to 10 mL with 0.2 mL\(^1\) solution of borate buffer (pH 10). Gentamicin, an aminoglycoside antibiotic, does not absorb UV light due to its weak chromophore, hence the need for derivatisation. The phthalaldehyde reagent was stored in amber-coloured bottles and kept in a dark cupboard prior to use. The reaction of the amine group in the aminoglycoside with the O-Phthalaldehyde in the presence of NaC yields a fluorescent isoindole which is measured at 332 nm absorbance.\(^12\) This method is superior to that used by Nnamani et al.,\(^1\) where mercaptoethanol, emits a characteristic unpleasant odour during the derivatization process.

**Preparation of wistar rat abdominal skin**

The hair of ether anesthetized Wistar rats weighing between 150-200 g was carefully removed with electric clippers, and the full thickness of skin was removed from the abdominal region. The epidermis was prepared surgically using a heat separation technique,\(^13\) which involved soaking the entire abdominal skin in water at 60°C for 45 s, followed by careful removal of the epidermis. The epidermis was washed three times with water and used for *ex vivo* permeability studies.

**Ex vivo permeation studies**

Permeation studies were performed using skin obtained from the rats (skin thickness 0.45-0.8 mm) mounted on modified Franz diffusion cells with a diffusion area of 3.71 cm\(^2\). The receptor compartment contained 30 mL phosphate buffer (pH of 7.4 at 37°C±0.2°C). One gram of each hydrogel formulation was applied on the skin surface in the donor compartment area with the stratum corneum facing downwards. An aliquot of 1 mL was withdrawn at predetermined time intervals and replaced with an equal volume of fresh media. The samples were analysed using a UV/visible spectrophotometer (UV-Vis 2600 Shimadzu Analytical and measuring instruments) after derivatisation using O-Phthalaldehyde reagent. Absorbance was measured at 332 nm. All experiments were performed in triplicate.

Cumulative amounts of permeated drug (µg/cm\(^2\)) was plotted against time in hours and drug (µg/cm\(^2\)/hr) at steady state was calculated by dividing the slope of the linear portion of the curve by the area of the exposed skin surface (3.71 cm\(^2\)). The permeation coefficient was deduced by dividing the flux by the area of the exposed skin surface (3.71 cm\(^2\)) and the time of exposure in hours.

The optimised gentamicin hydrogel derived from statistical analysis was then compared with a marketed gentamicin formulation via *ex vivo* permeation studies and the data obtained were evaluated using ANOVA followed by Tukey’s test at p<0.05.

**Ex vivo permeation kinetics of drug release**

The mechanism of drug release from the hydrogels was analysed by fitting the release data to various release kinetic models. The zero-order (Ko), first order (Kf), and Korsemeyer-Peppas model was used to determine the model with the best fit.\(^11\)

**Accelerated stability testing**

The International Council for Harmonisation of Technical Requirements for Pharmaceuticals for Human Use (ICH) guidelines (40°C/75% RH) were followed in the accelerated stability testing of the optimised hydrogel formulation. The hydrogels were packed in amber-coloured jars and kept in a stability chamber with a set temperature and relative humidity. The formulations were subjected to accelerated stability testing at both room temperature and at 40°C and parameters were recorded on day 0, 10, 15, 30, and 90. The formulations were evaluated for pH, assay, gel index (GI), and percentage of drug released at 12 hours.

The study was approved by the College of Medicine, University of Lagos Health Research Ethics Committee (CMULHREC number: CMUL/HREC/04/17/117, 15.04.2015).

**Statistical analysis**

The data were expressed mean standard deviation (±SD) using ANOVA (±SD). Significant differences (p<0.05) of mean values were determined using the Tukey test.

**RESULTS**

**FTIR spectroscopy**

The individual spectra and the physical mixture spectra were recorded and analysed. The fingerprint region and absorbance values relating to the relevant bioactive functional groups of the individual spectra analysed and the physical mixtures showed an absence of interaction between gentamicin, carbopol, and Transcutol and EETC as shown in Figure 1. Absorbance patterns corresponding in position and relative intensity to those in the FTIR spectra of the individual components were observed with no significant change in FTIR spectra after introduction of the polymers and the permeation enhancers, thus indicating a lack of physical or chemical interaction, as shown in Figure 1.

Some major bands/peaks on the spectra were 3089.75 cm\(^{-1}\) O-H stretching, 1706.88 cm\(^{-1}\) carboxyl group, which is characteristic of the principal absorption peaks of Carbopol\(^®\). Gentamicin was characterized by principal peaks at 610.22 cm\(^{-1}\) and 1100-1400 cm\(^{-1}\). The spectral data of EETC confirmed the presence of functional groups such as hydroxyl, an ester group and an aldehyde group among others 2920.21 cm\(^{-1}\) - C-H stretching depicting alkenes and aryl groups, 1730.32 cm\(^{-1}\) - aldehyde/ketone –C=O- stretching at 1440 cm\(^{-1}\). For the compatibility study, the FTIR spectra were compared and there was no disappearance or major shift of important peaks in the physical mixtures of Carbopol\(^®\), EETC, and Transcutol spectra.

**Gel characterization**

All formulations had a pale greenish colour and a good gel-like consistency. The hydrogels formulated using the CCD had
drug content variation form 94.5%-102.9% had excellent ex vivo permeation profiles. GeH 11 had 258.06 µg/cm²±0.43 of gentamicin permeated at eight hours in as seen in Figure 2. The pH of all the 14 hydrogel formulations ranged of 5.50-5.95 after neutralization with TEA as shown in Table 2. This pH range is important for use on the surface of wounds to facilitate wound healing at an acidic pH.1,3

Rheologic measurements
Spindle 4 was used for the viscometric characterisation of the hydrogels. Characterisation was performed at 20-60 rpm, which is the working range for this spindle. As the shear rate increased there was a corresponding decrease in the viscosity of the gels. This was evaluated exponentially using the Power Law as shown in Equation 3.

\[ T = K D^n \]  

where T is shear stress, K is GI or consistency index, D is shear rate, and n is flow index. Gel indices computed ranged from 1.02 to 2.11 as shown in Table 2.

Statistical modelling and analysis
Analysis of the experimental data using the Design Expert software revealed that the quadratic model was suitable for describing the formulation of the hydrogels. The final statistical models for predicting the flux and the amount of drug permeated after 12 hours are given in Equations 6 and 7.

\[ Y = 19.35 - 25.82X_1 - 0.044X_1^2 + 0.0097X_1X_2 + 11.86X_2 \]  

\[ Y = 31.50 - 189.67X_1 - 0.28X_2 + 1.29X_1X_2 + 123.55X_2^2 \]  

The values of the responses as predicted by Equations 4 and 5 are presented in Table 3 alongside the experimental data for comparison. The results of ANOVA conducted to determine the fit of the statistical models for flux and drug permeation are presented in Tables 3, 4, and 5.

Table 2. Observed responses in the central composite design for gentamicin hydrogels

| Hydrogel formulation | pH       | Assay (%) | PEc (cm/hr) | Gel index | Kinetic modelling |
|----------------------|----------|-----------|-------------|-----------|------------------|
|                      |          |           |             | Zero-order (Ko) | First-order (Kf) | Higuchi model (Kh) |
| GeH 1                | 5.67±0.05| 98.65±0.11| 1.77        | 1.02      | 0.912            | 0.567               | 0.943       |
| GeH 2                | 5.69±0.04| 98.73±0.28| 1.91        | 1.76      | 0.893            | 0.329               | 0.954       |
| GeH 3                | 5.78±0.32| 99.43±0.78| 1.72        | 1.89      | 0.903            | 0.510               | 0.964       |
| GeH 4                | 5.51±0.09| 101.02±0.21| 1.85       | 1.76      | 0.911            | 0.622               | 0.932       |
| GeH 5                | 5.86±0.03| 98.82±0.88| 1.70        | 1.65      | 0.954            | 0.476               | 0.954       |
| GeH 6                | 5.54±0.48| 99.72±0.01| 1.69        | 1.97      | 0.963            | 0.619               | 0.922       |
| GeH 7                | 5.88±0.11| 100.07±0.29| 1.73       | 1.56      | 0.945            | 0.409               | 0.973       |
| GeH 8                | 5.89±0.77| 98.99±0.89| 1.71        | 2.00      | 0.953            | 0.743               | 0.917       |
| GeH 9                | 5.61±0.37| 98.62±0.67| 1.77        | 1.97      | 0.944            | 0.599               | 0.972       |
| GeH 10               | 5.92±0.11| 99.89±0.32| 1.78        | 1.56      | 0.891            | 0.421               | 0.954       |
| GeH 11               | 5.93±0.07| 100.33±0.06| 1.93      | 2.11      | 0.932            | 0.511               | 0.911       |
| GeH 12               | 5.55±0.32| 101.77±0.03| 1.72       | 1.99      | 0.894            | 0.613               | 0.919       |
| GeH 13               | 5.63±0.04| 102.31±0.15| 1.73       | 1.34      | 0.843            | 0.412               | 0.963       |
| GeH 14               | 5.78±0.12| 100.99±0.45| 1.72       | 1.69      | 0.909            | 0.592               | 0.951       |
Tables 3, 4, 5, and 6 show the results of ANOVA conducted to determine the fit of the statistical models representing the flux and drug permeation after twelve hours. Tables 3 and 4 show that the models for flux and drug permeation were statistically significant with very low p values of 0.0001 and 0.0019, respectively. The single-effect model terms representing the effect of Transcutol:EETC ratio and stirring speed for both models (Equations 5 and 6) were significant indicating changes in the values of these variables could affect the flux and drug permeation.

Table 5 shows that the models for flux and drug permeation had high $R^2$ values of 0.90 and 0.82, respectively. The $R^2$ value indicates the degree to which a model is able to predict a response. The closer the $R^2$ value is to unity, the better the model can predict the response.14,15 The high $R^2$ values obtained for both models show that there was significant fit between the observed and predicted values of flux and drug permeation. Table 5 also shows that the SD of the observations was relatively small compared with the mean values of flux and drug permeation showing that there was very little dispersion about the mean for the data predicted by both models. The experimental runs were performed with high reliability and precision as seen from the relatively low values of coefficient of variation obtained for flux and drug permeation (8.75 and 12.27, respectively).16 The adequate precision values of both models were greater than four. This shows that the models had adequate signals and thus could be used to navigate the design space.17

Effect of independent variables on hydrogel formulation

Figures 3 and 4 are response surface plots showing the effect of Transcutol:EETC ratio and stirring speed on the flux and drug permeation.

### Table 3. Experimental design matrix for gentamicin hydrogel formulation

| Run   | Factors | Coded values | Actual values | Flux (µg/cm²/hr) | Drug permeation (µg/cm²) |
|-------|---------|--------------|---------------|------------------|-------------------------|
|       | GeH 1   | X<sub>1</sub> 1  | X<sub>2</sub> 1  | 0.902 162.43 | 6.91 6.37 124.43  102.10 |
|       | GeH 2   | X<sub>1</sub> 1  | X<sub>2</sub> -1 | 0.902 77.57  | 9.25 9.43 186.93 176.70 |
|       | GeH 3   | X<sub>1</sub> -1 | X<sub>2</sub> 0  | 1/3 120.00     | 12.75 13.64 222.4  249.07 |
|       | GeH 4   | X<sub>1</sub> -1 | X<sub>2</sub> -1 | 3/7 77.57     | 14.24 13.83 254.8  235.96 |
|       | GeH 5   | X<sub>1</sub> 0  | X<sub>2</sub> 0  | 2/3 120.00     | 9.8 9.34 178.02  175.02 |
|       | GeH 6   | X<sub>1</sub> 0  | X<sub>2</sub> -1 | 2/3 60.00      | 11.21 11.64 196.26 209.49 |
|       | GeH 7   | X<sub>1</sub> 0  | X<sub>2</sub> 0  | 2/3 120.00     | 8.57 9.34 163.78  175.02 |
|       | GeH 8   | X<sub>1</sub> 1  | X<sub>2</sub> 0  | 1 120.00       | 7.63 7.69 114.21 128.71 |
|       | GeH 9   | X<sub>1</sub> 1  | X<sub>2</sub> 0  | 2/3 180.00     | 5.59 7.03 110.21 140.55 |
|       | GeH 10  | X<sub>1</sub> 0  | X<sub>2</sub> 2/3 | 120.00 9.71 9.34 184.98 175.02 |
|       | GeH 11  | X<sub>1</sub> 0  | X<sub>2</sub> 3/7 | 162.43 11.51 10.38 243.99 213.06 |
|       | GeH 12  | X<sub>1</sub> 0  | X<sub>2</sub> 2/3 | 120.00 9.98 9.34 183.99 175.02 |
|       | GeH 13  | X<sub>1</sub> 0  | X<sub>2</sub> 2/3 | 120.00 9.82 9.34 176.43 175.02 |
|       | GeH 14  | X<sub>1</sub> 0  | X<sub>2</sub> 2/3 | 120.00 9.05 9.34 165.34 175.02 |
drug permeation of the hydrogels, respectively. Lower levels of stirring speed and transcutol:EETC ratio enhanced the flux of the formulated hydrogel as shown in Figure 3. This is evidenced by the fact that the flux increased with a reduction in stirring speed. This observation was recorded at all values of Transcutol:EETC ratio investigated. A similar trend was also observed for the Transcutol:EETC ratio for all values of stirring speed investigated.

DISCUSSION

Gentamicin is freely soluble in water (hydrophilic) and much of the drug was present in the aqueous phase of the formulations, loosely attached at or near the particle surface, because the more hydrophilic the substance, the weaker the interaction with particle surface, and eventually the compound could be localized in the surfactant layer.\(^1\) When more drug particles at the periphery of the particle surface eventually encounter the polymeric cross-linked gel-matrices, stabilization would occur.\(^18,19\) Maximisation of skin uptake and delivery of a drug that is hydrophilic such as gentamicin in a hydrogel would thus be affected by increased stirring speeds above 60-77 rpm. Figure 4 shows that the drug permeation after twelve hours reduced with an increase in stirring speed. This trend was, however, more significant at a higher ratio of Transcutol to EETC compared with when the lower ratio was used. An increase of stirring speed above this point will ensure decreased porosity of the polymeric system increasing entrapment of the Transcutol:EETC within the hydrophilic matrix due to excessively intense agitation during formulation. This may consequently result in decreased release rates as seen in Figure 2 and 4, inadvertently negatively influencing permeation of gentamicin through the skin as seen in Figure 4.\(^1\) A burst effect as result of increase stirring speed may also account for the decreased drug permeation with an increase in stirring speed as the drug is freed from the polymeric matrices and as such cannot be transported through the biologic membrane using transcutol:EETC. This effect will account for why there is a reduced flux at higher stirring speeds, as shown in Table 3, where the experimental values were closely correlated with the predicted responses.

EETC has been studied for toxicity and biocompatibility and has been seen to be nontoxic to and biocompatible with mammalian cell lines, thus informing its use in this formulation development.\(^4-6\) EETC is very high in antioxidants, which lower inflammatory markers and facilitate wound healing by promoting fibroblast migration. This combination of EETC with transcutol (2-(2-Ethoxyethoxy)ethanol), a chemical permeation enhancer, synergistically causes diffusional resistance of the stratum corneum, thereby increasing migration of gentamicin through the skin via increased solubility in the stratum corneum. Transcutol:EETC at high concentrations facilitates interaction with stratum corneum lipids to increase fluid into the skin producing increased flux, Pec, and ultimately drug permeation. There was an inverse relationship between the drug permeation after twelve hours and the transcutol:EETC ratio. This ensures that increased permeation occurs in the first
12 hours of hydrogel application. The flux obtained ranging from 9.05 to 14.42 µg/cm²/hr (accounting for release at the linear portion of the gentamicin permeation curve in Figure 1, which represents the first 4 hours of drug release and permeation) showed that GeH 4 with flux 14.42 µg/cm²/hr had the highest flux, which reflects increased permeation at an optimal stirring speed. This trend was observed both at high as well as at low values of stirring speed. However, the correlation between the drug permeation and transcutol:EETC ratio was more significant at high values of stirring speed due to the increased porosity of the hydrogel matrix. The mechanism of release predominantly observed was the Higuchi model, thus relating that the initial drug concentration in the hydrogel matrix was much higher than drug solubility with drug diffusion taking place in one dimension with edge effect being negligible, this accounts for increased release through pores in the matrix hydrogel system.

Optimisation of hydrogel formulation

Numerical optimization was performed to maximize the flux and drug permeation using the Design Expert software. The optimum conditions were chosen from the results obtained from the software possessing the highest desirability. These conditions are summarized in Table 6. The implication of these results is that the maximum flux and drug permeation can only be obtained if the independent variables are fixed at the values shown in Table 7. Accelerated stability testing of the optimized formulation showed that no variation in pH, assay, gel index, percentage of drug released at 12 hours was observed, as shown in Table 8.

Validation of statistical models

The validity of the statistical models used for predicting flux and drug permeation was confirmed by conducting three confirmation experimental runs at the identified optimum conditions (Table 6). The results showed that there was no significant difference between the experimental results and those predicted by the statistical models. The excellent correlation between the predicted and measured values shows the validity of the statistical models. Figure 5 shows the percentage of drug released from the optimised formulation in comparison with a marketed formulation. The ex vivo permeation study showed an improved release rate was obtained compared with the marketed topical formulation with 100% release occurring at 12 hours with a significant effect (p<0.05) compared with the marketed brand, which had 90% release at the same time point. This result is in consonance with the optimum value of drug permeation given in Table 7. Flux was obtained as 16.9 µg/cm²/hr compared with 9.98 µg/cm²/hr for the marketed formulation, and the amount of drug permeated after 12 hours was 260 µg/cm².

### Table 6. Statistical information for ANOVA

| Parameter     | Flux | Drug permeation |
|---------------|------|-----------------|
| R-squared     | 0.90 | 0.82            |
| Mean          | 9.72 | 178.98          |
| Standard deviation | 0.85 | 21.97          |
| C.V %         | 8.75 | 12.27           |
| Adeq. precision | 14.68 | 11.20        |

ANOVA: Analysis of variance

### Table 7. Summary of optimum conditions for the formulation of hydrogels

| Variables            | Optimum value |
|----------------------|---------------|
| Stirring speed (rpm) | 60            |
| Transcutol:EETC ratio| 1/3           |
| Flux (µg/cm²/hr)     | 16.1          |
| Drug permeation after 12 hours (µg/cm²) | 258 |

EETC: Ethanolic extract of Tetracarpidium conophorum

### Table 8. Accelerated stability testing on the optimized (GeH) gentamicin hydrogel at 40°C/75% RH (p≤0.05)

| Duration | pH     | Assay (%) | Gel index | Flux    | Drug permeation after 12 hours (µg/cm²) | Appearance and homogeneity |
|----------|--------|-----------|-----------|---------|----------------------------------------|---------------------------|
| Day 0    | 5.77±0.05 | 99.65±0.11 | 1.93±0.02 | 15.93±0.22 | 254.03±1.01                      | Satisfactory              |
| Day 10   | 5.79±0.04 | 99.73±0.28 | 1.96±0.03 | 15.83±0.74 | 251.72±2.11                      | Satisfactory              |
| Day 15   | 5.78±0.32 | 99.43±0.78 | 1.99±0.01 | 16.09±0.32 | 253.32±0.93                      | Satisfactory              |
| Day 30   | 5.78±0.07 | 100.02±0.21| 1.96±0.01 | 15.99±0.18 | 250.11±1.02                      | Satisfactory              |
| Day 90   | 5.76±0.03 | 99.98±0.03 | 1.93±0.04 | 15.93±0.07 | 254.2±0.97                       | Satisfactory              |

Figure 5 Comparison of percentage of gentamicin release from the optimised hydrogel formulation and marketed gentamicin topical formulation against time in hours
stressed conditions, as shown in Table 8, after accelerated gel formulations were stable at room temperature and under transdermal permeation within 12 hours at 60 rpm. All the gel formulations were stable at room temperature and under stressed conditions, as shown in Table 8, after accelerated stability testing.

CONCLUSIONS
Optimisation of gentamicin hydrogel using central composite statistical design is valid for the prediction of drug permeation and flux using variables in formulation preparation i.e. stirring speed and permeation enhancer:antioxidant ratio, thus showing their interaction with each other. The contour plots aided in the prediction of the value of transcutol:EETC and stirring speed, which would provide an optimized gentamicin hydrogel with optimal and drug permeation after 12 hours. The evaluation of therapeutic efficacy in an animal model is recommended for further studies.

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