Comparative phytochemical, antioxidant, and antibacterial study of different parts of Doigota plants (Bixa orellana L.)

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Abstract

Background: Bixa orellana L. is a plant native to Brazil but grows in other parts of the world including Bangladesh. This plant has been used for many years by indigenous people around the globe for a variety of applications. In Bangladesh, it is known as Doigota and the plant is underutilized and reported endangered. The present study aimed to perform comparative profiling where aqueous, ethanolic, and methanolic extracts of the seed, branch, and leaf of B. orellana plants were investigated for in vitro phytochemicals, antioxidant, and antibacterial activities.

Results: Phytochemical screening indicated the presence of nine different phytochemicals in varying degrees, which suggests the chemical diversity of this plant. Quantification of phenolics, flavonoids, and tannins was performed where leaf extracts contained significantly higher quantities of these phytochemicals than extracts from seed and branch. The highest phenolics (186.02 mg/g) were found in the methanolic extract of the leaf, whereas the highest flavonoids (61.01 mg/g) and tannins (36.74 mg/g) were found in aqueous extract of the leaf. Antioxidant activity has been evaluated by DPPH scavenging assay where the aqueous extract of leaf showed 93.62% scavenging activity at 800 μg/mL extract concentration. Although the extracts prevented the growth of both Gram-positive and Gram-negative bacteria, the antibacterial activity (zone of inhibition) of aqueous and methanolic extracts of the leaves were higher than those of the seed and branch extracts against Gram-negative (Escherichia coli) and Gram-positive (Bacillus subtilis) bacteria, respectively.

Conclusions: We found that B. orellana leaf extracts possess the highest antibacterial activities. The extracts had significant antioxidant and free radical-scavenging activities. The antioxidant and antibacterial activities could be attributed to high flavonoid and phenolic contents. These results suggest that the extracts of B. orellana possess bioactive compounds that could be used for different purposes such as source of new antibiotics or alternative natural preservatives in food matrices due to mainly their antioxidant and antibacterial activities.

Keywords: Antibiotics, Antioxidants, Bioactive compounds, Doigota, Phytochemicals
Background

*Bixa orellana* L., belonging to the family of Bixaceae, known as the annatto, achiote, or lipstick tree, is a small shrubby tree native of Central and Southern America (Giorgi et al. 2013). It is known as Doigota in Bengali and reported endangered (Rahman and Fakir 2015). It is found mainly in the Brazilian forest ecosystems with humid tropical conditions and now being cultivated in tropical areas throughout the world (Mello and Lima 1990). It has achieved economic importance because it contains natural reddish-orange pigment in its seeds composed primarily of the carotenoids bixin and norbixin, for which it is also referred to as annatto (Stohs 2014). It has been traditionally used by the people as a dye to paint hair and body as insect repellant or protectant from sunburn (Patani et al. 2013). Food industries use *B. orellana* as a colorant in butter, margarine, cheese, biscuits, dairy products, and chocolates (Shilpi et al. 2006). It is also implemented in textile industries for dyeing of jute, wool, cotton, and silk. In the global market, it has a preference next to saffron as a natural food colorant due to its non-toxicity as well as its tendency to preserve natural food flavor (Smith 2006). It has also traditional value in folklore medicine as it is consumed as a digestive aid and as a medication to control stomach disorders. Particularly, leaf extracts have been used for many years to treat headache, dysentery, fever, various microbial diseases, heartburn, and indigestion (Stohs 2014). The seeds are used to treat fever, skin diseases, and gonorrhea. The seed-derived oil is used to cure leprosy, and the decoction is used to treat jaundice (Tamil Selvi et al. 2011; Yusuf et al. 1994).

Plants have been a rich source of effective and safe medicine. About 80% of the world’s population is still dependent on traditional medicines. Different phytochemicals are present in the medicinal plants which are the key source of new pharmaceuticals and healthcare products (Tiwari 2008). The history of plants used for medicinal purposes is probably as old as human history. Plants have been subjected to extensive research because they contain secondary metabolites (bioactive components) that have beneficial therapeutic effects and various health-promoting activities. It has been reported that bioactive compounds such as phenolics, flavonoids, tannins, and carotenoids exhibit antioxidant properties (Rincón et al. 2014). In recent years, natural resources have been widely explored for the presence of antioxidants and efforts have been made to identify active antioxidant compounds from natural sources. Free radical scavenging antioxidants are important because they prevent various kinds of diseases and disorders (Ksouri et al. 2009).

Antibiotics are commonly used to treat infectious diseases caused by bacteria. There are various antibiotics available in the markets, which are being used against different pathogenic microorganisms. However, antibiotic resistance led to the emergence of multidrug-resistant bacteria which has become a major concern for public health. This phenomenon gives rise to the concept of using naturally derived compounds for their potential application to control pathogenic microorganisms. Natural compounds may offer a new source of antibiotics more effective than synthetic antibiotics. Regarding this issue, over the last few years, a good number of studies have been conducted worldwide. Previously, three studies involving *B. orellana* were performed in Bangladesh so far. In the light of previous findings, we designed an elaborate and comparative approach to analyze three different parts (seed, branch, and leaf) of *B. orellana*, which were extracted with three different solvents (water, ethanol, and methanol). The analysis covered preliminary phytochemical screening, quantitative estimation of phenols, flavonoids, and tannins; antioxidant activity by 2,2-diphenyl-1-picrylhydrazyl (DPPH) scavenging assay; and antibacterial analysis using both Gram-negative (*Escherichia coli*) and Gram-positive (*Bacillus subtilis*) bacteria. The study was mainly aimed to explore the comparative phytochemical profile and the bioactivity and also the consistency with previous findings.

Materials and methods

**Chemicals and reagents**

All chemicals and reagents used in this study were of analytical grade. Aluminum chloride, (+)-catechin, potassium iodide, DPPH (2,2-diphenyl-1-picrylhydrazyl), and gallic acid were purchased from Sigma Aldrich, USA. Glacial acetic acid was purchased from Scharlau, Spain. Tannic acid was purchased from Loba Chemie, India. Ethanol, methanol, Fehling A and B, sodium hydroxide, and sodium nitrite were bought from Merck, Germany. Folin-Ciocalteu reagent was purchased from SRL, India. Sodium carbonate, chloroform, and iodine were bought from Alfa Aesar, UK. Hexane and ethyl acetate were bought from Daejang Chemicals and Duksun Pure Chemicals, South Korea, respectively.

**Study location and plant materials**

The experiments were conducted at the Cell Genetics and Plant Biotechnology Laboratory, Department of Biotechnology and Genetic Engineering, Jahangirnagar University, Dhaka-1342, Bangladesh (23° 53′ 14” N 90°15′ 56” E). Initially, the seeds of Doigota (*B. orellana*) were collected from some cultivated plants in a garden located at Hemayetpur, Dhaka, Bangladesh, during November 2016. The seeds were germinated, and plants were cultivated in the laboratory garden with appropriate conditions and care during January 2017. Seeds, branches, and leaves were collected during March 2019.
from these 2 years old cultivated plants and used as the experimental materials in this study.

Preparation of extracts
The plant materials (seed, branch cutting, and leaf) were sun dried for 7 days and then dried in a hot air oven (JSR, Korea) at 50 °C for 7 days. The dried plant materials were powdered in a mechanical grinder. In the extraction process, distilled water, 70% ethanol, and 70% methanol were used as solvents. The powdered materials (10 g) were taken in conical flasks with 70% ethanol, 70% methanol, and distilled water and kept in an orbital shaker for 3 days at room temperature. The extracts were filtered through Whatman No. 1 filter paper, and the filtrates were concentrated using a rotary evaporator at 45 °C. Finally, stock solutions of extracts (10 mg/mL) were prepared using 0.1 N NaCl. For qualitative screening and quantitative determination of phytochemicals, the stock extracts were diluted to the concentration of 1 mg/mL.

Qualitative screening of phytochemicals
The freshly prepared crude extracts were qualitatively tested for the presence of secondary metabolites such as carbohydrates, alkaloids, glycosides, coumarins, tannins, phenols, flavonoids, saponins, and proteins by following the methods described in our previous literature (Ahmed et al. 2018).

Carbohydrates
One milliliter of different extracts was taken into test tubes to which equal volume of Fehling’s A and Fehling’s B were added. The tubes were heated at 65 °C in a water bath for 10–15 min. Redbrick precipitate indicated the presence of carbohydrates (Ahmed et al. 2018).

Alkaloids
One milliliter of different extracts was taken into test tubes, and 4–6 drops of Wagner’s reagent [1.27 g of iodine and 2 g of potassium iodide in 100 mL of water] was added. The radish brown precipitate indicated the presence of alkaloids (Rizk 1982).

Glycosides
One milliliter of different extracts was mixed with 1 mL of glacial acetic acid in test tubes. Then, 5–6 drops of 1% ferric chloride solution were added. Brown color ring produced at the top indicated the presence of glycoside (Peash et al. 2017).

Coumarins
One milliliter of the extract was added with 1.5 mL of 10% NaOH. The chemical reaction produced yellow color and indicated the presence of coumarin (Ugochukwu et al. 2013).

Tannins
One milliliter of extracts was mixed with 1 mL of distilled water, and then, 4–5 drops of 1% ferric chloride were added. Blue and the greenish-black color indicated the presence of gallic tannin and catecholic tannin respectively (Peash et al. 2017).

Phenols
One milliliter of ethanol was added to 1 mL of extracts. Then, 6–7 drops of 1% ferric chloride solution were added to each tube. Formation of the green or blue or purple color indicated the presence of phenol (Soloway and Wilen 1952).

Flavonoids
Three to four drops of 20% NaOH solution were added to 2 mL of extracts. The intense yellow color formed and become colorless when 4–5 drops of dilute HCl was added. This indicated the presence of flavonoids (Ugochukwu et al. 2013).

Saponins
Two milliliters of distilled water was added to 1 mL of extracts and shaken for 5 min. The presence of 1-cm-thick foam for 10 min indicated the presence of saponin (Kumar et al. 2009).

Proteins
One milliliter of the extract was treated with few drops of concentrated HNO₃. The formation of yellow color indicated the presence of proteins (Yadav et al. 2017).

Quantitative determination of phytochemicals

Determination of total phenolic content
The total phenolic content of the extracts was estimated according to the method described previously (Shohael et al. 2006a). 0.1 mL aliquots of extracts and standards of different concentrations were mixed with 2.5 mL deionized water followed by the addition of 0.1 mL (2 N) Folin–Ciocalteu reagent. They were mixed well and allowed to stand for 6 min before 0.5 mL of a 20% sodium carbonate solution was added. The color developed after 30 min at room temperature. The absorbance was measured at 760 nm in a UV-visible spectrophotometer (T60 UV-Visible Spectrophotometer, PG Instruments Ltd., UK). The concentration of total phenol was determined using gallic acid as standard. Total phenolic content was expressed as milligrams of gallic acid equivalent (mg GAE)/g of extract.
**Determination of total flavonoid content**

Total flavonoid content was determined following the method described previously (Shohael et al. 2006b). Briefly, 0.25 mL of extracts or (+)-catechin standard solution was mixed with 1.25 mL of distilled water, followed by the addition of 0.75 mL of a 5% sodium nitrite solution. After 6 min, 0.15 mL of a 10% aluminum chloride solution was added, and the mixture was allowed to stand for a further 5 min and then 0.5 mL of 1 M sodium hydroxide was added. The mixture was brought to 2.5 mL with distilled water and mixed well. The absorbance was measured immediately at 510 nm in a T60 UV-visible spectrophotometer. The concentration of total flavonoids was determined using (+)-catechin as standard. The concentration of flavonoids was expressed as milligrams of catechin equivalent (mg CE)/g of extract.

**Determination of total tannin content**

Total tannin content was determined using the Folin-Ciocalteu reagent as described previously (Ahmed et al. 2018). Briefly, 0.1 mL of the sample extract is added with 7.5 mL of distilled water. Then, 0.5 mL of Folin-Ciocalteu reagent and 1 mL of 35% sodium carbonate solution were added. The total volume was adjusted to 10 mL by adding distilled water. The mixture was incubated at room temperature for 30 min, and the absorbance was measured at 725 nm in a T60 UV-visible spectrophotometer. Blank was prepared with water instead of the sample. A set of standard solutions of tannic acid was read against a blank. The concentration of total tannin was determined using tannic acid as standard. The concentration of tannin was expressed as milligrams of tannic acid equivalent (mg TAE)/g of extract.

**DPPH free radical scavenging activity**

DPPH (2,2-diphenyl-1-picrylhydrazyl) free radical scavenging activity was carried out using the method described previously (Kabir et al. 2016). Different concentrations (800, 400, 200, 100, 50 μg/mL) of extract were dissolved in methanol. Three milliliters of a 0.004% DPPH solution was added to each test tube. The mixture was incubated for 30 min in dark condition. The absorbance was measured at 517 nm in a T60 UV-visible spectrophotometer. The percent inhibition activity was calculated from \( \frac{(A_C - A_S)}{A_C} \times 100 \), where \( A_C \) is the absorbance of the control and \( A_S \) is the absorbance of the sample.

**Antibacterial assay**

In vitro antibacterial assay was performed by using the disc diffusion method. Pure cultures of bacteria isolated from clinical specimens were obtained from the Department of Microbiology, Jahangirnagar University. The organisms were Gram-negative *Escherichia coli* and Gram-positive *Bacillus subtilis*. The pure culture contains the test organisms on solid media. They were transferred to the sterile nutrient broth and incubated for 18 h at 37 °C in an incubator (HYSC, Korea). The optical density of the culture was determined using a T60 UV-visible spectrophotometer at 600 nm. An inoculating loop was used to aseptically transfer the bacterial strains into 5 mL of media, which was used as a starter culture. The Whatman No. 1 filter paper was used for the preparation of discs which were 5 mm in diameter. The disks were impregnated with different concentrations (10 mg/mL, 7.5 mg/mL, 5 mg/mL, and 2.5 mg/mL) of extracts. Each piece of the disc was impregnated with 50 μL of extract. So the final concentrations of extracts on the discs were 500, 375, 250, and 125 μg/disc. The prepared working discs were completely air-dried inside of the class II biological safety cabinet (ESCO, Singapore) and placed on the nutrient agar medium plate. The nutrient agar medium plate was spread with 100 μL of starter bacterial culture before the placement of the disks. The culture plates were incubated at 37 °C in an incubator (HYSC, Korea). After incubation, they were examined for the inhibition of growth. The growth inhibition was measured in terms of zones of inhibition. Zones of inhibition were measured in millimeter-scale using a digital slide caliper. Oxoid™ Streptomycin antimicrobial susceptibility discs (25 μg) (ThermoFisher Scientific, USA) were used as positive control, and sterile distilled water was used as the negative control.

**Statistical analysis**

Significant differences among mean values were compared by Tukey’s honestly significant difference (HSD) test at a level of significance of \( P < 0.05 \). All data were displayed as the mean ± standard error of the mean (SEM) at least three independent biological replications. The statistical analysis was performed using Statistical Package for Social Science software (SPSS, version 16.0, IBM Corporation, NY).

**Results**

**Qualitative screening of phytochemicals**

Phytochemicals are active chemical compounds produced by plants and involved in several functions. The qualitative phytochemical analysis gave an overview of what secondary metabolites or bioactive components are present in the extracts. Preliminary phytochemical screening was conducted for nine different chemical compounds. Table 1 summarizes the results of the preliminary phytochemical screening. These data showed that the extracts were rich in all the phytochemicals tested. All three of the extracts obtained from seeds indicated that nine different phytochemicals were present in
the samples. Extracts from branch cuttings showed some negative results which indicated some phytochemicals were not detected in particular extract. Such as carbohydrates were not detected in aqueous and ethanolic extracts of the branch, glycosides were not detected in ethanolic extracts of the branch; saponins were not detected in aqueous and ethanolic extracts of the branch. Ethanolic and methanolic extracts of branch showed the very good presence of tannins and phenols. For the leaf, in aqueous extract, the presence of alkaloids, glycosides, and coumarins was not detected. All three extracts of leaf were rich in tannins and phenols, contained proteins in a moderate amount, and carbohydrates were detected in a lower amount.

**Quantitative determination of phytochemicals**

**Determination of total phenolic content**

Total phenolic content in different extracts obtained from seed, branch, and leaf of *B. orellana* is shown in Fig. 1. The highest (186.02 mg/g of extract) phenolic content was found in the methanolic extract of leaf, and the lowest (46.34 mg/g of extract) phenolic content was found in ethanolic extract of seed.

**Determination of total flavonoid content**

Figure 2 represents the total flavonoid content in different extracts obtained from seed, branch, and leaf of *B. orellana*. The highest (61.01 mg/g of extract) flavonoid content was found in aqueous extract of leaf, and the lowest (6.84 mg/g of extract) flavonoid content was found in methanolic extract of seed.

**Determination of total tannin content**

Total tannin content in different extracts obtained from seed, branch, and leaf of *B. orellana* is shown in Fig. 3. The highest (36.74 mg/g of extract) tannin content was found in aqueous extract of leaf, and the lowest (5.86

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**Table 1 Preliminary phytochemical screening of different extracts of *B. orellana***

| Compounds   | Aqueous extract (seed) | Ethanolic extract (seed) | Methanolic extract (seed) | Aqueous extract (branch) | Ethanolic extract (branch) | Methanolic extract (branch) | Aqueous extract (leaf) | Ethanolic extract (leaf) | Methanolic extract (leaf) |
|-------------|------------------------|--------------------------|---------------------------|--------------------------|---------------------------|---------------------------|-----------------------|-------------------------|---------------------------|
| Carbohydrates | ++                     | ++                       | +++                       | +                        | ++                        | +++                       | +                     | +                       | +                         |
| Alkaloids    | ++                     | ++                       | +++                       | +                        | ++                        | +++                       | +                     | +                       | +                         |
| Glycosides   | ++                     | ++                       | +++                       | +                        | ++                        | +++                       | +                     | +                       | +                         |
| Coumarins    | +                      | ++                       | +                         | +                        | ++                        | +++                       | +                     | +                       | +                         |
| Tannins      | +                      | ++                       | +                         | +                        | ++                        | +++                       | +                     | +                       | +                         |
| Phenols      | +                      | ++                       | +++                       | +                        | ++                        | +++                       | +                     | +                       | +                         |
| Flavonoids   | ++                     | ++                       | +++                       | +                        | ++                        | +++                       | +                     | +                       | +                         |
| Saponins     | +++                    | +++                      | +++                       | +                        | ++                        | +++                       | +                     | +                       | +                         |
| Proteins     | ++                     | ++                       | +++                       | +                        | ++                        | +++                       | +                     | +                       | +                         |

Indication: +++ = high; ++ = moderate; + = low; − = not detected
mg/g of extract) tannin content was found in methanolic extract of the branch.

**DPPH radical scavenging activity**

DPPH radical scavenging activity was used to evaluate the antioxidant potential of different extracts from *B. orellana*. Figure 4 showed the comparative antioxidant activity in terms of DPPH radical scavenging assay where the antioxidant potentials were expressed as % DPPH scavenged. From the assay, it was clearly evident that the extracts obtained from leaf were best for DPPH scavenging. The highest activity was found for aqueous extract of the leaf at 800 μg/mL extract concentration, which showed 93.62% scavenging activity followed by ethanolic extract of the leaf (92.47%) and methanolic extract of the leaf (91.72%) at 800 μg/mL extract concentration. For the extracts obtained from the branch, it was seen that ethanolic extract (88.9%) and methanolic extract (80.37%) had better scavenging potential at 800 μg/mL extract concentration than the aqueous extract (68.76%). And the extracts obtained from seed had lower antioxidant capacity where we have found 53.05, 49.28, and 42.60% DPPH scavenging activity at 800 μg/mL extract concentration of methanolic, ethanolic, and aqueous extract respectively.
Antibacterial activity

Antibacterial activity was performed for both Gram-negative (Escherichia coli) and Gram-positive (Bacillus subtilis) bacteria. Tables 2 and 3 accommodate the zone of inhibition observed for Gram-negative bacteria E. coli and Gram-positive bacteria B. subtilis by different concentrations of extracts, respectively. Inhibition of bacterial growth was observed for both the bacterial species by all of the extracts. For Gram-negative bacteria E. coli, the highest zone of inhibition (13.1 mm) was found for aqueous extract of leaf and the lowest zone of inhibition (6.1 mm) was found for the methanolic extract of the branch at 10 mg/mL extract concentration.

For Gram-positive bacteria B. subtilis, the highest zone of inhibition (15.7 mm) was found for the methanolic extract of leaf and the lowest zone of inhibition (7.9 mm) was found for the methanolic extract of seed at 10 mg/mL extract concentration.

Discussion

The presence of a wide range of phytochemical constituents including carbohydrate, alkaloid, glycosides,

Table 2 Zone of inhibition observed for Gram-negative bacteria Escherichia coli by different concentrations of extracts

| Samples                | Zone of inhibition (mm) |
|------------------------|-------------------------|
|                        | 10 mg/mL | 7.5 mg/mL | 5 mg/mL | 2.5 mg/mL |
| Aqueous extract (seed) | 9.1 ± 0.3 | 8.5 ± 0.3 | 8.1 ± 0.3 | 7.6 ± 0.3 |
| Ethanol extract (seed) | 8.5 ± 0.3 | 8.3 ± 0.2 | 8.1 ± 0.2 | 7.8 ± 0.2 |
| Methanol extract (seed)| 6.9 ± 0.2 | 6.7 ± 0.2 | 5.9 ± 0.2 | 5.4 ± 0.2 |
| Aqueous extract (branch)| 10.2 ± 0.3 | 8.9 ± 0.2 | 8.5 ± 0.3 | 7.2 ± 0.2 |
| Ethanol extract (branch)| 9.2 ± 0.2 | 8.8 ± 0.2 | 7.6 ± 0.2 | 7.1 ± 0.3 |
| Methanol extract (branch)| 6.1 ± 0.2 | 5.7 ± 0.2 | 0 | 0 |
| Aqueous extract (leaf) | 13.1 ± 0.3 | 9.0 ± 0.3 | 8.2 ± 0.3 | 7.4 ± 0.3 |
| Ethanol extract (leaf) | 12.6 ± 0.3 | 9.2 ± 0.2 | 8.0 ± 0.2 | 7.1 ± 0.4 |
| Methanol extract (leaf) | 12.9 ± 0.2 | 8.4 ± 0.3 | 7.8 ± 0.2 | 7.2 ± 0.2 |
than 4000 compounds (Ren et al. 2003). They have been
phenolic compounds in plants, which constitute more
methanol than the ethanol and water. B. orellana
pounds present in
The result of our study suggested the phenolic com-
cessions of
Zarza-García et al. 2017). Our study showed that leaf extracts
and also supported by previous studies. The present
study also showed that the seed and branch samples
content as 113.53, 158.77, and 163.00 mg/g in
different solvents were rich in phenolic compounds.
Phenolics are prevalent in leaves as evident by our study
also supported by previous studies. The present
study also showed that the seed and branch samples
contain a considerable quantity of phenolic compounds.
A study performed in Colombia by Viuda-Martos et al.
(2012) utilized ethanolic extract of leaf and seed to
quantify the phenolics and flavonoids and reported total
phenol content as 1.81 mg/g for ethanolic leaf extract
and 0.74 mg/g for ethanolic seed extract (Viuda-Martos
et al. 2012). The study by Zarza-García et al. (2017)
demonstrated a study in Mexico by utilizing three Mayan ac-
cessions of B. orellana leaves where they found total
phenolic content as 113.53, 158.77, and 163.00 mg/g in
three accessions respectively (Zarza-García et al. 2017).
The result of our study suggested the phenolic com-
ounds present in B. orellana leaf are more soluble in
methanol than the ethanol and water.

Flavonoids resemble a class of naturally occurring
phenolic compounds in plants, which constitute more
than 4000 compounds (Ren et al. 2003). They have been
shown to exert a variety of biological activities. Traditional
folk medicine utilizes flavonoids as an active con-
stituent and has been used by physicians to manage
various diseases and disorders (Havsteen 1983). In our
study, the total flavonoid content is considerably higher
in extracts obtained from the leaf. The study by Viuda-
Martos et al. (2012) found similar result where the total
flavonoid content was found as 65.53 mg/g for ethanolic
leaf extract and 22.18 mg/g for ethanolic seed extract
(Viuda-Martos et al. 2012).

Tannins are natural, water-soluble, polyphenolic com-
ounds with a molecular weight ranging from 500 to
3000 Da (Dalton). Different plant species were shown to
contain tannins that have specific physical and chemical
properties as well as outstanding pharmacological sig-
ificance (Chung et al. 1998). In our study, as the other
two phytochemicals quantified, leaf extracts are also rich
in tannins than the branch or seed. Karmakar et al.
(2018) quantify tannins in the ethanolic extract of seeds
of B. orellana, where total tannin content was 25 mg/g
of dry extract (Karmakar et al. 2018). Both the flavonoid
and tannin are prevalent in the aqueous extract which
suggested that the flavonoid and tannin present in B.
orellana leaf are more soluble in water than the ethanol
and methanol.

Plants contain different antioxidants naturally which
are responsible for inhibiting the harmful effects of oxida-
tive stress. The antioxidant activities are attributed to
the presence of active phytoconstituents, including pheno-
lcohols, flavonoids, and related compounds (Subramanian
et al. 2013). In our study, the free radical scavenging ac-
tivity was evaluated by the DPPH (2,2-diphenyl-1-picryl-
hydrazyl) method, which is regarded as an easy, reliable,
sensitive, and rapid method, and almost 90% of antioxi-
dant studies are conducted with this method (Moon and
Shibamoto 2009). Shilpi et al. (2006) found that 500 μg/
ml concentration of leaf extract exhibited 90.58% DPPH
scavenging activity, whereas Karmakar et al. (2018) uti-
lized ethanolic extract of seeds and found 75.77% DPPH

| Samples                  | Zone of inhibition (mm)                     |
|--------------------------|---------------------------------------------|
|                          | 10 mg/mL | 7.5 mg/mL | 5 mg/mL | 2.5 mg/mL |
| Aqueous extract (seed)    | 10.1 ± 0.2 | 9.3 ± 0.3 | 8.2 ± 0.2 | 8.0 ± 0.2 |
| Ethanol extract (seed)    | 13.0 ± 0.2 | 11.3 ± 0.2 | 10.0 ± 0.3 | 8.2 ± 0.2 |
| Methanol extract (seed)   | 7.9 ± 0.2  | 7.5 ± 0.2  | 7.3 ± 0.3  | 7.1 ± 0.2 |
| Aqueous extract (branch)  | 8.6 ± 0.3  | 8.1 ± 0.3  | 7.8 ± 0.2  | 6.5 ± 0.2 |
| Ethanol extract (branch)  | 10.9 ± 0.2 | 9.5 ± 0.2  | 8.3 ± 0.3  | 7.4 ± 0.3 |
| Methanol extract (branch) | 9.8 ± 0.3  | 8.8 ± 0.3  | 8.4 ± 0.2  | 7.1 ± 0.2 |
| Aqueous extract (leaf)    | 12.4 ± 0.2 | 9.3 ± 0.2  | 8.7 ± 0.2  | 8.0 ± 0.3 |
| Ethanol extract (leaf)    | 12.9 ± 0.3 | 9.6 ± 0.2  | 8.5 ± 0.4  | 7.3 ± 0.2 |
| Methanol extract (leaf)   | 15.7 ± 0.2 | 10.8 ± 0.3 | 8.0 ± 0.3  | 7.0 ± 0.2 |
scavenging activity at 512 μg/mL concentration. Viuda-Martos et al. (2012) reported 94.20% and 93.01% DPPH scavenging activity for ethanolic leaf and seed extracts respectively.

In the antibacterial assay, higher activity was found by extracts obtained from leaf than the extracts from seed and branch. In the case of leaf extracts, activity was higher against Gram-positive bacteria than Gram-negative bacteria. However, the extracts obtained from seeds and stems were also effective to some extent against the bacteria, which suggested their utility as antibacterial agent. The difference in cell wall structure between Gram-positive and Gram-negative bacteria is a contributing factor to the susceptibility of bacteria to phytochemicals. In Gram-negative bacteria, the cell wall is more complex due to the presence of a lipopolysaccharide layer outside the peptidoglycan layer. That is why the layer may turn out to be impermeable to many substances, whereas gram-positive bacteria have only the peptidoglycan layer. That is why the permeability is higher for some substances. In light of this explanation, we can say that the phytochemicals present are more active against the Gram-positive bacteria because they have better permeability to penetrate into the cell structure. The previous study by Abayomi et al. demonstrated the positive bactericidal activities of B. orellana seed extracts against Gram-positive bacteria Staphylococcus aureus and Bacillus subtilis. However, they did not find any activity against Gram-negative bacteria such as Escherichia coli (Abayomi et al. 2014). The study performed by Irobi et al. (1996) indicated significant antimicrobial activity by B. orellana leaf extracts against standard strains of Gram-positive bacteria including B. subtilis, Streptococcus faecalis, and S. aureus. However, their study found slight inhibition for the Gram-negative bacteria (Irobi et al. 1996). Fleischer et al. (2003) utilized ethanolic extract of leaf and seed to study the antibacterial activity of B. orellana where they found activity against all the Gram-positive and Gram-negative bacteria (Fleischer et al. 2003). A Bangladeshi study performed by Alim et al. (2016) showed the antibacterial activity of methanolic extract of B. orellana leaf where the higher zone of inhibition was found for Gram-negative bacteria (E. coli) than the Gram-positive bacteria (B. subtilis) (Alim et al. 2016). Another Bangladeshi study by Shilpi et al. (2006) showed that methanolic extract of the leaf has inhibitory activity against Gram-negative bacteria (E. coli) (Shilpi et al. 2006). Karmakar et al. (2018) utilized ethanolic extract of seeds where they found antibacterial activity against Gram-negative bacteria (E. coli) (Karmakar et al. 2018). The study by Shilpi et al. (2006) and Karmakar et al. (2018) did not use B. subtilis in their experiment.

**Conclusion**

From the present study, it is quite clear that among the three different plant parts tested, the leaves are most useful in every aspect as they are rich in phytochemicals and antioxidants as well as they showed the best antibacterial activity. However, the solvents used in this study did not let us get a concise conclusion for which solvent is best because they showed a different pattern in different experiments. But we can say that aqueous extract is more effective followed by methanolic and ethanolic extracts. It is quite clear from the study that the concentration and type of bioactive substances present in the extracts depended on several factors which may be attributed to the season, geography and environment, soil type and climate, processing methods such as type of solvent, etc. The present study has updated the state of knowledge about B. orellana to some extent by bringing together the existing literature information and novelties such as comparative phytochemical profiles, antioxidant activities, and antibacterial potentials. Moreover, comparative insights into phytochemical profiles may help to reveal the type and nature of chemical diversity in this useful plant. It is also worth mentioning that the study was performed from the perspective of Bangladesh, where there is a report of the plant is at risk of being lost in all or part of their distribution ranges because of reduction in their population numbers due to habitat destruction and over-collecting. The study may be very preliminary in nature, but it is the foundation for future investigation. A more detailed investigation is required to explore the in-depth medicinal properties as well as their mechanism of action.

**Abbreviations**

DPPH: 2,2-Diphenyl-1-picrylhydrazyl; GAE: Gallic acid equivalent; CE: Catechin equivalent; TAE: Tannic acid equivalent; TPC: Total phenolic content; TFC: Total flavonoid content; TTC: Total tannin content; AES: Aqueous extract (seed); EES: Ethanolic extract (seed); MES: Methanolic extract (seed); AEB: Aqueous extract (branch); EEB: Ethanolic extract (branch); MEB: Methanolic extract (branch); AEL: Aqueous extract (leaf); EEL: Ethanolic extract (leaf); MEL: Methanolic extract (leaf); SEM: Standard error of the mean; SD: Standard deviation

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**Authors’ contributions**

Abdullah Mohammad Shohael generated the idea and designed the experiments. Sium Ahmed, Biklis Mehrin Moni, and Shawon Ahmed collected and prepared the samples and performed all the experiments. Sium Ahmed, Biklis Mehrin Moni, and Shawon Ahmed analyzed the data. Sium Ahmed wrote the manuscript. Biklis Mehrin Moni, Donald James Gomes, and Abdullah Mohammad Shohael reviewed the manuscript. The authors read and approved the final manuscript.
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**Availability of data and materials**
All data generated during this study are included in this published article.

**Ethics approval and consent to participate**
Not applicable.

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Not applicable.

**Competing interests**
The authors declare that they have no competing interests.

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