Mechanically Enhanced Self-Stratified Acrylic/Silicone Antifouling Coatings

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Abstract: Great attention has been paid to silicone-based fouling-release coatings (FRCs) in the realm of maritime antifouling due to their highly efficient and eco-friendly properties, but many challenges remain for developing a silicone-based FRC that improves its adhesion performance without reducing the antifouling property. Herein, a non-toxic silicone-based FRC has been developed by integrating acrylic resin (AR) with a silicon resin (PDMS) to spontaneously form a self-stratified AR/PDMS coating. The AR/PDMS antifouling coating still has the same fouling-release performance but improved adhesion strength (from 0.4 to 2.0 MPa) in comparison with pristine PDMS. Moreover, the antifouling coating has proven to be extremely stable in different environments (such as pH, heating, and ultraviolet exposure). The study provides a facile and convenient self-stratified strategy to develop antifouling coatings, contributing to environmentally friendly coatings in marine applications.

Keywords: antibiofouling; biofouling resistance; self-stratifying; adhesion force

1. Introduction

Marine biofouling can dramatically increase economic and environmental burdens, becoming a worldwide challenge in the realm of the maritime industry [1–5]. According to the America Defense Science and Technology Agency, the estimated total cost associated with hull fouling can reach to ~260 million per annum in their Navy fleet [6,7]. Marine biofouling is formed by the undesirable accumulation of marine microorganisms, animals, and plants on submerged surfaces causing significant extra fuel consumption and higher emissions of greenhouse gases from marine ships. For marine facilities, biofouling causes and accelerates the corrosion of metal structures, thus greatly increasing maintenance costs [8–10]. Antifouling coatings are the most economic, convenient, and effective approach to preventing the accumulation of marine organisms and metal corrosion. The primary coatings use biocide-containing materials to inhibit their adhesion or kill biofouling organisms [11–13]. Although effective, the use of these paints is increasingly restricted because of their toxicity, which simultaneously affects other non-targeted species [14,15]. Hence, there is a pressing need for developing environmentally friendly and effective antifouling systems.

To date, eco-friendly antifouling strategies, such as fouling-resistant coatings and fouling-release coatings (FRCs), have been developed to combat marine organisms. Fouling-resistant coatings can prevent the adhesion of biofouling. They are usually composed of hydrophilic polymers, including those based on zwitterions and poly (ethylene glycol) (PEG) [16]. However, they are generally swollen in ocean conditions, which results in poor mechanical properties and limited applications. Thus, FRCs with highly efficient and eco-friendly properties are particularly attractive in marine antifouling applications [17–19]. Among those, as one of the most common FRCs, poly(dimethylsiloxane) (PDMS) possesses...
both a low elastic modulus and low surface energy, on which the organisms only weakly adhere to the surfaces and can be readily removed from the PDMS surface by gentle mechanical cleaning or the shear forces of water flow, without the involvement of any chemical/biological antifoulants [20–24]. On the other hand, the low elastic modulus and low surface energy also endow the PDMS coating with a low adhesion force to regular substrates. Generally, although the adhesion force between the coating and substrate can be largely enhanced, the introduced coupling agent inevitably increases the coating elasticity modulus and surface energy, hence heavily compromising the surface fouling-release performance [25,26]. Therefore, it is still challenging to propose an effective strategy that improves its adhesion performance without reducing the fouling-release property.

Currently, the addition of a bonding layer has been reported to achieve good adhesion ability without reducing the antifouling property [11]. However, this approach is time-consuming and requires complex procedures and environmental waste generation, which is not always adapted to industrial requirements [27]. As a solution to these short-comings, a new economical and effective self-stratifying system has been developed, allowing a one-step formation of thermodynamically stable multi-layer coating structures after the mixture of two incompatible polymers is applied to a substrate. Accordingly, the system provides not only decreased processing time and economic benefits, but also an optimized surface and good adhesion properties for the coating [28–32]. Moreover, the preferential distribution of concentration through the film thickness greatly eliminates interfacial adhesion failure without compromising the advantages of a multi-layer system [33,34]. In this study, a commercially available acrylic resin (AR) has been considered as the alternative resin in a self-stratified system. ARs serve as reliable anticorrosion materials due to their resistance against adverse environmental conditions and excellent adhesion ability.

Herein, we prepared a novel self-stratified acrylic/PDMS fouling-releasing coating, together with the merits of easy preparation, non-toxicity and a high adhesion on the substrate. Benefiting from its low surface energy, the PDMS segments in the coating tend to be enriched on the outmost layer of the surface, contributing to active fouling-releasing behaviors. Meanwhile, the acrylic components, with relatively high surface energy, readily settle down to the substrate to enhance interfacial adhesion. Moreover, the resultant self-migration coating can be applied on varied substrates because of its excellent adhesion ability, unique mechanical strengths and facile procedure. Both the adhesive properties of the coating and fouling-release performances were systematically explored.

2. Materials and Methods

2.1. Materials

Two industrially available resins were selected: an acrylic resin purchased from Guangzhou Huakui Chemical Co., Ltd. (Guangzhou, China) and a silicone resin (184 PDMS elastomer) purchased from Dow Corning Corporation (Shanghai, China). Bacterial strain *Paracoccus pantotrophus* ATCC 35,512 (*P. pantotrophus*) was obtained from Chuanxiang Biotechnology, Ltd. (Shanghai, China). Butyl acetate (≥ 99.5%) was purchased from Sigma-Aldrich (St. Louis, MO, USA) and used as received. Commercially available carbon steel panels (Q 235, 20 mm × 20 mm × 2 mm) and glass panels were used as the substrates. The carbon steel sheets (C, 0.2%; Mn, 1.2%; and Si, 0.35%) were mechanically polished with emery paper of different grades (600, 900 and 1200). The glass panels were prepared as squares with the dimensions of 25 mm × 25 mm × 2 mm. The samples were cleaned with acetone and distilled water and dried with nitrogen flow.

2.2. Stratified Coating Preparation

Each resin was dissolved evenly in butyl acetate solvent, with an optimized concentration of 50% (w/v, polymer/solvent). Then, different contents of PDMS solutions (30, 50, and 70% relative to the acrylic resin weight) were dispersed uniformly into acrylic resin solutions, respectively, and thoroughly stirred by magnetic stirring at 40 °C for 20 min. The PDMS curing agent (10%, relative to the PDMS weight) was added to the incompatible
resin system and mixed for 3 min. The coating suspensions were applied separately on the polished steel substrates and glass substrates by drop-casting. Then, the coating was dried at ambient temperature for 24 h to avoid an abrupt curing, then cured for 2 h at 110 °C in an oven.

2.3. Adhesion Force Tests

According to the ISO 4624:2016 standards, the pull-off adhesion tests of the coating on a steel substrate were carried out with the aid of an Instron 1122 tensile tester. The adhesion of the coating was evaluated by the greatest tensile strength that it could bear at the moment of detachment. Additionally, the adhesion of the coatings was further determined according to the ISO-2409:2013 standard cross-cut test. Briefly, cell grids containing 6 parallel and vertical crosscuts were drawn out on a coating surface with a space of 3 mm between grid blades. Then, 3M-600 adhesive tape (0.44 N/mm) was firmly placed on the affected area and followed by quick removal. Finally, the adhesion level was confirmed by the ratio of the remaining coating area to the original coating area, and the larger ratio demonstrated the higher adhesion level.

2.4. Coating Stability Test

To further evaluate the stability of the AR-PDMS$_{0.5}$ coating, the coating was immersed in aqueous solutions with different pH values (pH = 1, 7, 10). The influence of the thermal aging treatment and UV irradiation on the surface performance of the AR-PDMS$_{0.5}$ coating was also investigated.

2.5. Mimetic Measurement of Barnacle Adhesion Strength

The mimetic measurement of barnacle adhesion strength was used to evaluate the adhesion of marine biofouling on the samples. According to ASTM D5618 (2011), an aluminum stud (diameter 10 mm, height 10 mm) was stuck on the coatings by epoxy adhesive. The adhesive was cured for 24 h at 110 °C in an oven. The maximal shear force was measured by a force gauge (SUNDOO SH-500, Wenzhou Shandu Instrument Co., Ltd., Wenzhou, China).

2.6. Antibacterial Adhesion Test

*P. pantotrophus* was employed as the model bacteria. Bacterial cells were grown in the LB culture medium, and then collected by centrifugation and washed with phosphate buffered saline (PBS, pH = 7.3). Finally, the bacteria were resuspended in PBS to a cell density of $10^6$ CFU mL$^{-1}$. The sterilized acrylic/silicone samples (1 cm × 1 cm) were added by 1 mL bacterial suspension. After incubation at 37 °C for 12 h with a shaker (130 r min$^{-1}$), the sample surfaces were gently washed by dipping in PBS three times (changed to clean PBS for each step), then placed in fresh medium after 3 min ultrasonic treatment. Moreover, the number of attached cells was determined by a drop plate assay.

2.7. Other Characterization

The contact angles (CAs) were measured by a drop shape analysis instrument (DSA, KRÜSSGMBH, Hamburg 100, Shanghai, China). Young’s modulus and hardness data were collected with a Nanoindenter (Hysitron Ti-900 Tribolndenter, Hysitron, Shanghai, China). Scanning electron microscopy (SEM, Hitachi SU8020, Hitachi, Guangzhou, China) was used for the detection of the layer interfacial connection and coating thickness. The difference in chemical compositions between the top and bottom coating surface was characterized by cross-section EDS analysis and attenuated total reflectance Fourier-transform infrared spectroscopy (ATR-FTIR, Bruker Vertex 70, Bruker, Shanghai, China).
3. Results and Discussion

3.1. Characterization of Acrylic/Silicone Coating

A novel class of dynamic antifouling coating was fabricated on the carbon steel substrate, which was composed of acrylic resin and silicon resin. Figure 1 illustrates the preparation process of the antifouling coating (AR/PDMS) by a two-phase mixture applied on the substrate. The self-stratifying approach allows the one-step formation of thermodynamically stable two-layer structures, combining optimized surface and adhesion properties. The PDMS elastomers are generally prepared by a hydrosilylation reaction (Figure S1). To examine the PDMS migration to the coating surface, their static contact angles were measured (Figure 2a). The probe liquids we used included water, soybean oil and hexadecane with different surface tensions. It is worth noting that the static CAs on the stratified coating surface were very close to the CA values on a pure silicone surface, whatever the probe liquids used, indicating that the PDMS segments with low surface energy migrated to the coating surface. Moreover, the WCA of the AR-PDMS0.7 surface (~117°) was approximately equal to that of pure PDMS (~118°), showing that AR-PDMS0.7 coatings possessed high hydrophobicity (Figure 2b). It has been reported that hydrophobicity can make the coating stable during long-term seawater immersion [35].

Figure 1. Schematic illustration of the fabrication of antifouling coating.

![Schematic illustration of the fabrication of antifouling coating.](image)

Figure 2. (a) Contact angles of various probe liquids on various samples (AR, AR-PDMS0.3, AR-PDMS0.5, AR-PDMS0.7 and PDMS). (b) Photographic images of water droplets on various samples (AR, AR-PDMS0.5, AR-PDMS0.7 and PDMS). (c) ATR-FTIR spectra of PDMS, AR, top layer and base layer of AR-PDMS0.5. (d) Young’s modulus and hardness of the samples.
ATR-FTIR was used to examine the top layer and base layer of the detached AR-PDMS$_x$ coating from the substrate. As for the top layer on both AR-PDMS$_{0.5}$ surfaces (Figure 2c), the band at 1010 cm$^{-1}$ was assigned to the Si–O stretch in Si–O–Si. The peak at 1258 cm$^{-1}$ was assigned to symmetric –CH$_3$ deformation within ≡Si–CH$_3$ moieties. These characteristic peaks of PDMS confirmed that the coating surface was mainly composed of PDMS components [36]. For the spectrum of the bottom layer, the peak at 1725 cm$^{-1}$ represented the C=O stretch [37], which were characteristic peaks of acrylic resin. It could be concluded that the acrylic resin remains at the bottom of the film during the process.

Since the elasticity plays an important role in the FR performance of the coatings, the elasticity of the coatings was investigated. Figure 2d shows the Young’s modulus and hardness of the coatings. The pure AR coating showed a high Young’s modulus of ~5790 MPa and a very high hardness of ~5788 MPa. As for the coating of AR-PDMS$_{0.3}$, a relatively high value of Young’s modulus (~4.7 MPa) was observed. As PDMS content increased, the results showed a declining trend owing to the migration of PDMS to the coating surface and the high flexibility of PDMS. The AR-PDMS$_{0.5}$ and AR-PDMS$_{0.7}$ possessed relatively low Young’s moduli (~2.6 and ~2.5 MPa, respectively), as compared to those of the control PDMS (~2.1 MPa), indicating that a high elasticity was maintained. A similar trend in coating hardness was also observed. The coating AR-PDMS$_{0.3}$ with the lowest PDMS content showed the highest value in hardness (~4.4 MPa). The increased PDMS contents caused significant decreases in coating hardness, with values of ~1.8 and 1.7 MPa for the samples of AR-PDMS$_{0.5}$ and AR-PDMS$_{0.7}$, which are comparable to the PDMS control (~1.6 MPa).

3.2. Self-Stratification of Acrylic/Silicone Coating

As shown in Figure 3a, the typical stratification pattern of AR-PDMS$_{0.5}$ with a thickness of ~230 µm, was clearly evidenced. Notably, the layers were well-adhered, without any detectable interlayer adhesion failure, indirectly indicating that the stratified coating is mechanically stable and hard to peel between the two layers [33]. To further confirm the composition of the coating (top and bottom layers), the EDS mappings were carried out. The silicon mapping demonstrated that the top layer of the coating was mostly composed of PDMS content, while the carbon mapping indicated that the bottom layer mainly consisted of acrylic resin components (Figure 3b,c), which is consistent with the results observed in the FTIR spectra. We further used EDS line scanning to examine the axial distribution of silicon and carbon along the acrylic/silicone coating (Figure 3d). Notably, the line scan analysis revealed that the gradual increase in silicon content was accompanied by a decrease in carbon content at the interface of the two-layer system (scanning distance: 38–45 µm). The formation of gradient structures may be due to the faster curing rate of the coating than the phase separation rate, resulting in the coating curing before complete separation [28]. This result is critical for us to explain the reason why the coating possessed the mechanically enhanced performance as listed in the following section.
Robust adhesion is a significant precondition for coating applications. Here, a pull-off test was adopted to investigate the adhesion behaviors of between the AR-PDMS$_x$ coating and the corresponding metal substrates. As shown in Figure 4a, the PDMS control exhibited a weak adhesion strength (~0.4 MPa) due to its nonpolar chemical nature, making it easy to detach from the metal substrate. In sharp contrast, as the presence of the acrylic component increased, the adhesion strength of AR-PDMS$_x$ coatings showed much higher adhesion strength, caused by the migration of acrylic resin to the substrate. Although the pull-off strength of AR-PDMS$_x$ declined to a slight degree compared to the pure acrylic coating, the adhesion force still showed a much higher value to ~2.00 MPa, as compared to the PDMS control. To explore whether the stratified coating can be peeled off between the top and bottom layers, an analysis of the interfacial adhesion properties was conducted. The AR-PDMS$_x$ exhibited a higher interfacial adhesion strength than traditional coatings with a two-step painting process (Figure 4b). The two-layer coating showed an adhesion strength of ~0.6 MPa. In contrast, the AR-PDMS$_{0.7}$ and AR-PDMS$_{0.5}$ coatings showed greater interfacial adhesion strength: ~1.5 MPa and 1.2 MPa, respectively. This is probably because curing agents may enhance the compatibility of two resin combinations and could provide efficient bonding between layers [28]. We further evaluated AR-PDMS$_{0.5}$ coating adhesion by a cross-cut tape test. The photographs shown in Figure S2 demonstrated that AR-PDMS$_{0.5}$ exhibited a superior adhesion to metal substrates after the scratch and tape-peeling process, compared to pure PDMS coating. The edges of the sample cuts were intact without any coating detachment, demonstrating that AR-PDMS$_{0.5}$ was classified to grade 0 (excellent adhesion strength). Moreover, the excellent adhesion performance made the coating stable during long-term artificial seawater immersion (Figure S3). Additionally, we examined the adhesion performance of AR-PDMS$_x$ coatings to the glass base, which also showed an outstanding adhesive force (Figure S4).
112.4° ± 1.8°. Clearly, the antifouling coating of AR-PDMS0.5 was nicely preserved against sandpaper abrasion.

3.4. Coating Stability and Abrasion Resistance

The marine environments vary greatly in different areas regarding temperature, pH, and other factors. Therefore, the influences of the acid–base immersion, UV irradiation and thermal aging treatment on the surface properties of the AR-PDMS0.5 coating were investigated. The AR-PDMS0.5 was immersed in an acid (pH = 4), alkali (pH = 10) and aqueous solution, respectively, for different incubation times ranging from 0 to 15 days. Figure 4c shows that the WCAs had virtually no change on the surface after being immersed in three solutions for 15 days, indicating that the coating possessed excellent pH stability. Moreover, Figure 4d shows the results of UV aging resistance, obtained by exposing the AR-PDMS0.5 to a UV lamp (365 nm, 5 mW/cm²). Upon UV exposure for 15 days, WCAs remained constant at 2.1° ± 0.2°. Subsequently, the thermal aging test was performed by heat treatment at 100 °C for 15 days (Figure 4e). Within this interval, the CAs remained nearly constant over time, demonstrating a superior high-temperature resistance. These results indicated that the AR-PDMS0.5 coating possessed a reliable durability, which could bear such harsh conditions such as different pH values, UV irradiation and heating treatment.

Mechanical robustness is also a key property of coatings in practical applications; therefore, a rigorous method was used to further study the mechanical stability of the AR-PDMS0.5 coating. The AR-PDMS0.5 coatings were placed on a 2000-mesh sandpaper, then a 50 g weight was set up above the sample. Each cycle included a rightward thrust motion, followed by a leftward thrust motion of the coating on 2000-mesh sandpaper, with a wear distance of 10 cm. Figure 4f shows that the WACs on the AR-PDMS0.5 surface exhibited minimal change with the increasing number of cycles of sandpaper abrasion. After 100 cycles of sand abrasion, the CA of the coating slightly decreased from 115.4° ± 2.9° to 112.4° ± 1.8°. Clearly, the antifouling coating of AR-PDMS0.5 was nicely preserved against sandpaper abrasion.
3.5. Antifouling Property

To evaluate the antifouling ability of the AR-PDMS$_{0.5}$ coating, the anti-smudge performance of the AR-PDMS$_{0.5}$ against ink was investigated. In Figure 5a, the ink formed distinct marks on the AR coating, and the ink stain was not removed by wiping with a cotton ball. By contrast, the ink on the PDMS and AR-PDMS$_{0.5}$ surfaces contracted into weak patchy marks. Moreover, the ink trace was readily wiped off with a cotton ball. These results showed that the AR-PDMS$_{0.5}$ coating had a good antifouling performance when soaked in ink.

Moreover, the submerged surface of the hull will inevitably contact and adhere to the sludge in the seawater. Thus, the coatings were tested for their ability to resist sludge adhesion by natural sludge suspension. As shown in Figure 5b, after 12 h of exposure to natural sludge suspension, compared to pristine AR coatings, there was basically no sludge adhesion on the AR-PDMS$_{0.5}$ surface, exhibiting excellent anti-sludge adhesion properties. We also applied the AR-PDMS$_x$ coating on glass substrates (Figure S5). The excellent anti-smudge property of the coating toward the glass substrate was attributed to the migration of PDMS to the coating surface.

3.6. Antibacterial Adhesion Performance

Previous reports have shown that bacterial adhesion is a pivotal step in biofilm formation [38]. Therefore, *P. pantotrophus* was used as the representative bacteria to test the antibacterial adhesion of AR-PDMS$_x$ coatings. As shown in Figure 6a,b, after 12 h of exposure in bacterial suspension, high-density *P. pantotrophus* bacterial colonies ($2.5 \times 10^5$ cfu/cm$^2$) on the AR coating were observed. In comparison, very few *P. pantotrophus* bacterial colonies could be found on the PDMS ($5.2 \times 10^4$ cfu/cm$^2$), PDMS-AR$_{0.5}$ ($5.3 \times 10^4$ cfu/cm$^2$), and PDMS-AR$_{0.7}$ ($5.4 \times 10^4$ cfu/cm$^2$) coatings, indicating that those coatings exhibited ~80% reduction in bacterial adhesion.

![Figure 5. Photographs of a series of antifouling tests (a) an oil-based permanent ink marker on the sample coatings. (b) exposing in an aqueous suspension of sludge for 12 h.](image_url)
The mimetic pseudo barnacle adhesion is widely used to evaluate the barnacle adhesion strength in the lab [35]. As shown in Figure 6c, the removal strength (RS) of AR-PDMS$_x$ coating surfaces was significantly decreased (nearly ~84% reduction) compared to that of control AR (~1.80 MPa). Both AR-PDMS$_{0.5}$ and AR-PDMS$_{0.7}$ displayed a largely decreased removal strength, ranging from ~0.29 to ~0.28 MPa, which was almost comparable to that of the PDMS control (~0.25 MPa). It could be expected that the attached barnacles on the AR-PDMS$_x$ should be easily removed by the hydrodynamic forces at a low ship speed in a marine environment. Considering the performances of both mechanical stability and adhesion strength together, the AR-PDMS$_{0.5}$ was considered as the optimized candidate for potential antifouling applications.

4. Conclusions

In summary, a novel AR/PDMS self-stratified coating was developed for the first time with the merits of easy preparation and non-toxicity. This self-stratifying effect was conducive to the low surface energy characteristics of the coating surface and the formation of good adhesion to the substrate. The coating exhibited remarkable anti-smudge properties as well as antibiofouling performance, owing to the surface migration of the low surface energy PDMS elastic segments. The resultant coating showed a remarkably improved adhesion strength to the substrate (2.0 MPa), as compared to a pure PDMS coating (0.4 MPa). Moreover, this coating also demonstrated a robust stability, which can ensure normal functioning under extreme conditions such as acid/alkali solutions, heating and UV exposure. Overall, a self-stratified AR/PDMS fouling-releasing coating might provide a new method that can effectively overcome the challenges of biofouling in the field of maritime applications.

Supplementary Materials: The following supporting information can be downloaded at: https://www.mdpi.com/article/10.3390/coatings12020232/s1, Figure S1: Hydrosilylation reaction for the preparation of silicone elastomers, Figure S2: Cross-cut tape test results of (a) PDMS and (b) AR-PDMS$_{0.5}$, Figure S3: Photographic images of AR-PDMS$_{0.5}$ coatings after immersion in ASW for 0, 10, and 30 d, Figure S4: The adhesion strength of AR-PDMS$_x$ coatings to the glass substrate,
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