Research Article

A Facile Synthesis of New 2-Amino-4H-pyran-3-carbonitriles by a One-Pot Reaction of α, α’-Bis(arylidene) Cycloalkanones and Malononitrile in the Presence of K2CO3

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A rapid and environmentally friendly method is developed for the synthesis of a series of new substituted 2-amino-4H-pyran-3-carbonitriles through a one-pot condensation of malononitrile and α, α’-bis(arylidene) cycloalkanones in ethanol by using K2CO3 as a catalyst. Short experimental reaction times, excellent yields, no need to use cumbersome apparatus for purification of the products, and inexpensiveness and commercially availability of the catalyst are the advantages of this method.

1. Introduction

In the past few decades, the synthesis of new heterocyclic compounds has been a subject of great interest due to the wide applicability of them. Heterocyclic compounds occur very widely in nature and are essential to life. The importance of multicomponent reactions in organic synthesis has been recognized, and considerable efforts have been focused on the design and development of one-pot procedures for the generation of libraries of heterocyclic compounds [1, 2]. 4H-Pyran-3-carbonitriles and their derivatives are of considerable interest due to their pharmacological activities [3], such as spasmolytic, diuretic, anticoagulant, anticancer, and antianaphylactic activity [4–6]. Moreover, 4H-pyran-3-carbonitriles are useful intermediates for synthesis of various compounds, such as pyranopyridine derivatives [7], polyazanaphthalenes [8], pyranopyrimidines [9], and pyridin-2-ones [10].

Furthermore, 4H-pyran-3-carbonitriles represent building blocks of a series of natural products [11, 12]. A number of 2-amino-4H-pyran-3-carbonitriles are used as photoactive materials [13], pigments [14], and potential biodegradable agrochemicals [15], and consequently, numerous methods have been reported for the synthesis of these compounds. Thus, the synthesis of 4H-pyran is of much importance to organic chemists. Several methods have been reported for the synthesis of pyran derivatives via a three-component condensation of β-dicarbonyl compounds with aldehydes and malononitrile [16]. From the literature, we observed that very few catalysts have been used for the synthesis of 2-amino-4H-pyran-3-carbonitriles base on the reactions of α, α’-bis(arylidene) cycloalkanones with malononitrile, for example, NaOH/piperidine [17], KF-Al2O3 [18], and hexadecyltrimethyl ammonium bromide (HTMAB) [19]. However, these methods show varying degrees of success as well as limitations such as prolonged reaction times, low yields, and use of toxic solvents. Thus, the development of an alternate milder and clean procedure is highly demanding for the synthesis of 2-amino-4H-pyran-3-carbonitriles, which surpasses those limitations. Herein, we planned to synthesis of these compounds using sequential reactions of α, α’-bis(arylidene) cycloalkanones and malononitrile in the presence of K2CO3 as a catalyst in ethanol under reflux conditions (Scheme 1).

Nowadays, organic reactions in ethanol without the use of harmful organic solvents have attracted much attention, because ethanol is a cheap, safe, and environmentally benign
solvent [7]. In recent years, K2CO3 has been considered as an efficient, inexpensive, and readily available catalyst for several organic transformations [20, 21].

2. Results and Discussion

In continuation of our studies on the development of organic reactions [22–24], herein we report a highly versatile and efficient synthesis of 2-amino-4H-pyran-3-carbonitriles 3a–q (Scheme 1) from α,α′-bis(Arylidene) cycloalkanone 1, malononitrile 2 and catalytic amounts of K2CO3. In a typical reaction, a mixture of 1 and 2 (1 : 1) equivalents, respectively, and K2CO3 (cat.) was refluxed in ethanol for 5–60 min. The results are summarized in (Table 1).

The formation of the compounds 3 was assumed to proceed via formation of a Michael adduct intermediate followed by cyclization according to Scheme 2. A α,α′-bis(arylidene) cycloalkanones 1 were firstly condensed with malononitrile 2 to afford the intermediate 4, this step can be regarded as a Michael addition. Then, the intermediate 5 cyclized by nucleophilic attack of the OH group on the cyano (CN) moiety and gave the intermediate 6. Finally, the expected products 3 were afforded (Scheme 2) [17–19].

To test the catalysts, the reaction of α,α′-bis(arylidene) cyclohexanone and malononitrile in ethanol was selected as a model reaction. The scope and the generality of the present method were then further demonstrated by the reaction of various α,α′-bis(arylidene) cycloalkanones with malononitrile and K2CO3. In all cases, good yields with good selectivity were obtained. The catalyst plays a crucial role in the success of the reaction in terms of the rate and the yields. The present methodology afforded high yields of the products within short times (5–60 min). The results (Table 1, entries 1–17) indicated that substrates 1 bearing both electron-donating groups (such as alkoxy and methyl) and electron-withdrawing groups (such as halide) can be involved in this one-pot synthesis to afford desired products 3 with high yields. Thus, it should be concluded that the electronic nature of the substituents has no significant effect on this reaction.

In order to show the merits of K2CO3 over other catalysts reported in the literature, results for the synthesis of 2-amino-4H-pyran-3-carbonitriles obtained using K2CO3 as the catalyst were compared with those obtained using other catalysts. Table 2 clearly shows that K2CO3 appears to promote the reaction more effectively than a number of other catalysts, particularly in terms of the time and yield required to complete the reaction.

3. Conclusion

In conclusion, the present method is a simple and environmentally friendly procedure for the synthesis of a series of new 2-amino-4H-pyran-3-carbonitriles using catalytic amount of K2CO3. The simple experimental procedure, short reaction times, excellent yields of products, mild reaction condition, easy purification, economic availability of the catalyst, and green standard are the advantages of this method.

4. Method

α,α′-Bis(arylidene)cycloalkanones have been synthesized through cross-aldol condensation of cycloalkanones and aldehydes using our reported method [25].

4.1. General Procedure for Synthesis of 2-Amino-4H-pyran-3-carbonitrile Derivatives 3a–q. A mixture of appropriate α,α′-bis(arylidene)cycloalkanone 1 (1 mmol), malononitrile 2 (1 mmol) and 5% mol K2CO3 (0.05 mmol, 0.006 g) in ethanol 96% (10 mL) was refluxed for the appropriate time indicated in Table 1 (5–60 min). The progress of the reaction was monitored by TLC. After completion of the reaction, the reaction mixture was cooled to room temperature, and the resulting cream precipitate was filtered and washed with n-hexane (10 mL) to furnish the corresponding 2-amino-4H-pyran-3-carbonitriles.

The structure of the products was deduced from their IR, 1H NMR, 13C NMR, and elemental analysis. The spectral (IR, 1H NMR, 13C NMR) and analytical data of unknown compounds are given below.

4.1.1. 8-(4-fluorobenzylidene)-2-amino-4-(4-fluorophenyl)-5,6,7,8-tetrahydro-4H-chromene-3-carbonitrile (Entry 7, 3g).

Cream powder, IR(KBr): 3465, 3342, 2945, 2196, 1671, 1643, 1599, 1503, 1414, 1221, 1134, 834, 803 cm\(^{-1}\); 1HNMR (250 MHz, CDCl3): δ = 1.59–1.63 (m, 2H, CH2), 1.85–2.04 (m, 2H, CH2), 2.50–2.71 (m, 2H, CH2), 3.95 (s, 1H, CH), 4.55 (s, 2H, NH2), 6.82 (s, 1H, =CH), 6.99–7.07 (m, 4H, ArH); 13CNMR (62.9 MHz, CDCl3): δ = 22.17, 26.96, 27.37, 42.88, 60.44, 115.03, 115.36.
Table 1: Synthesis of 2-amino-4H-pyran-3-carbonitriles 3a–q.

| Entry | Z       | R       | Product | Time (min) | Yield (%)b | mp (°C)     | Ref |
|-------|---------|---------|---------|------------|------------|-------------|-----|
| 1     | CH₂     | H       | 3a      | 45         | 87         | 227–228     | [19]|
| 2     | CH₂     | 2-Cl    | 3b      | 5          | 90         | 213–214     | [19]|
| 3     | CH₂     | 2,4-Cl₂ | 3c      | 15         | 93         | 238–239     | [18]|
| 4     | CH₂-CH₂ | H       | 3d      | 10         | 95         | 228–230     |     |
| 5     | CH₂-CH₂ | 2-Cl    | 3e      | 10         | 85         | 237–238     | [19]|
| 6     | CH₂-CH₂ | 4-Cl    | 3f      | 15         | 85         | 215–216     | [19]|
| 7     | CH₂-CH₂ | 4-F     | 3g      | 10         | 90         | 222–224     |     |
| 8     | CH₂-CH₂ | 4-Br    | 3h      | 15         | 88         | 214–217     |     |
| 9     | CH₂-CH₂ | 4-Me    | 3i      | 60         | 90         | 161–162     | [19]|
| 10    | CH₂-CH₂ | 4-OMe   | 3j      | 10         | 80         | 220–222     |     |
| 11    | CH₂-CH₂ | 2,4-Cl₂ | 3k      | 15         | 87         | 195–196     | [18]|
| 12    | CH₂-CH₂ | 2-Cl, 6-F| 3l      | 10         | 85         | 233–236     |     |
| 13    | CH(CH₃)CH₂ | H       | 3m      | 20         | 90         | 199–202     |     |
| 14    | CH(CH₃)CH₂ | 2-Cl    | 3n      | 25         | 87         | 198–201     |     |
| 15    | CH(CH₃)CH₂ | 4-Cl    | 3o      | 15         | 85         | 208–209     |     |
| 16    | CH(CH₃)CH₂ | 4-Me    | 3p      | 60         | 75         | 214–218     |     |
| 17    | CH(CH₃)CH₂ | 4-OMe   | 3q      | 20         | 80         | 199–202     |     |

aReaction conditions: α,α′-bis(arylidene) cycloalkanones 1 (1 mmol), malononitrile 2 (1 mmol), K₂CO₃ (0.05 mmol, 5 mol%), EtOH (10 mL), reflux.
bIsolated yields.

4.1.2. 8-((4-bromobenzylidene)-2-amino-4-(4-bromophenyl)-5,6,7,8-tetrahydro-4H-chromene-3-carbonitrile (Entry 8, 3h).
Cream powder, IR(KBr): 3443, 3318, 3212, 2925, 1815, 1600, 1508, 1404, 1201, 1025, 831 cm⁻¹; ¹HNMR (250 MHz, CDCl₃): δ = 1.61–1.79 (m, 2H, CH₂), 1.84–2.16 (m, 2H, CH₂), 2.49–2.94 (m, 2H, CH₂), 3.93 (s, 1H, CH), 4.56 (s, 2H, NH₂), 6.79 (s, 1H, CH₂), 7.16–7.23 (m, 4H, ArH), 7.45–7.48 (m, 4H, ArH); ¹³CNMR (62.9 MHz, CDCl₃): δ = 22.11, 27.0, 27.37, 43.14, 60.03, 115.50, 115.84, 119.77, 121.37, 129.35, 129.48, 130.75, 130.88, 138.61, 141.40, 158.08, 151.80. Anal. Calcd For C₂₃H₁₈Br₂N₂O: C, 55.45; H, 3.64; N, 5.62; Found: C, 55.34; H, 3.60; N, 5.59.

4.1.3. 8-((4-methoxybenzylidene)-2-amino-4-(4-methoxyphenyl)-5,6,7,8-tetrahydro-4H-chromene-3-carbonitrile (Entry 10, 3j).
Cream powder, IR(KBr): 3446, 3335, 2925, 2836, 2188, 1667, 1630, 1603, 1508, 1404, 1249, 1127, 1029, 831 cm⁻¹; ¹HNMR (250 MHz, CDCl₃): δ = 1.34–1.62 (m, 2H, CH₂), 1.94–1.96 (m, 2H, CH₂), 2.53–2.91 (m, 2H, CH₂), 3.79 (s, 3H, OCH₃), 3.82 (s, 3H, OCH₃), 3.91 (s, 1H, CH), 4.51 (s, 1H, CH₂), 6.81 (s, 1H, CH₂), 6.85–6.91 (m, 4H, CH₂), 7.16–7.23 (m, 4H, ArH), 7.45–7.48 (m, 4H, ArH); ¹³CNMR (62.9 MHz, CDCl₃): δ = 22.11, 27.0, 27.37, 43.14.
Table 2: Comparison of results using K₂CO₃ with other catalyst for synthesis of 2-amino-4H-pyran-3-carbonitriles.

| Entry | Catalyst | Solvent | T  | Time | Yield (%) | Ref |
|-------|----------|---------|----|------|-----------|-----|
| 1     | NaOH/Piperidine | EtOH | MW | 5–9h | 70–71 | [17] |
| 2     | KF-AI₂O₃ | DMF | RT | 10–14h | 68–90 | [18] |
| 3     | HTMAB | H₂O | 110°C | 8h | 76–93 | [19] |
| 4     | K₂CO₃ | EtOH | Reflux | 5–60 min | 75–95 | This work |

ArH), 7.14–7.26 (m, 4H, ArH); ¹³CNMR (62.9 MHz, CDCl₃): δ = 22.99, 27.12, 27.35, 27.40, 42.76, 55.27, 60.81, 113.66, 114.10, 114.74, 122.02, 127.95, 128.92, 129.60, 129.70, 130.54, 135.10, 135.15, 141.42, 158.41, 158.81. Anal. Calcd For C₂₅H₂₃N₂O₂: C, 74.98; H, 6.04; N, 7.0; Found: C, 75.01; H, 6.07; N, 7.04.

4.1.4. 8-(2-chloro-6-fluorobenzylidene)-2-amino-4-(2-chloro-6-fluorophenyl)-5,6,7,8-tetrahydro-4H-chromene-3-carbonitrile (Entry 12, 3i). Cream powder, IR(KBr): 3456, 3339, 2944, 2913, 2188, 1672, 1637, 1597, 1443, 1413, 1240, 1130, 897, 780, 756 cm⁻¹; ¹HNMR (250 MHz, CDCl₃): δ = 1.59–1.65 (m, 2H, CH₂), 1.93–2.11 (m, 2H, CH₂), 2.25–2.86 (m, 2H, CH₂), 4.60 (s, 1H, CH), 4.89 (s, 2H, NH₂), 6.56 (s, 1H, =CH), 6.84–7.03 (m, 2H, ArH), 7.14–7.25 (m, 4H, ArH); ¹³CNMR (62.9 MHz, CDCl₃): δ = 21.80, 27.15, 27.56, 43.10, 58.01, 113.26, 113.89, 114.27, 119.60, 124.11, 124.39, 124.76, 125.10, 128.76, 128.91, 129.17, 132.33, 132.56, 133.53, 134.72, 135.05, 158.41, 160.13. Anal. Calcd For C₂₅H₁₈Cl₁₂F₂N₂O: C, 62.04; H, 3.62; N, 6.29; Found: C, 62.08; H, 3.64; N, 2.23.

4.1.5. 2-amino-8-benzylidene-6-methyl-4-phenyl-5,6,7,8-tetrahydro-4H-chromene-3-carbonitrile (Entry 13, 3m). Cream powder, IR(KBr): 3433, 3329, 2925, 2910, 2190, 1670, 1632, 1594, 1485, 1409, 1009 cm⁻¹; ¹HNMR (250 MHz, CDCl₃): δ = 0.90 (d, 3H, J = 6.2 Hz, CH₃), 1.60–2.27 (m, 4H, 2CH₂), 2.81–2.87 (m, 1H, CH), 3.94 (s, 1H, CH), 4.49 (s, 2H, NH₂), 6.88 (s, 1H, =CH), 7.22–7.37 (m, 10H, ArH); ¹³CNMR (62.9 MHz, CDCl₃): δ = 21.00, 28.55, 34.76, 35.09, 43.09, 61.01, 114.62, 119.20, 120.00, 122.86, 126.82, 127.36, 127.90, 128.22, 128.82, 129.27, 137.01, 142.13, 143.01, 158.93. Anal. Calcd For C₂₅H₂₅N₂O: C, 81.33; H, 6.26; N, 7.90; Found: C, 81.40; H, 6.23; N, 7.95.

4.1.6. 8-(2-chlorobenzylidene)-2-amino-4-(2-chlorophenyl)-6-methyl-5,6,7,8-tetrahydro-4H-chromene-3-carbonitrile (Entry 14, 3n). Cream powder, IR(KBr): 3472, 3330, 2945, 2924, 2192, 1674, 1635, 1594, 1411, 1130, 1036, 738 cm⁻¹; ¹HNMR (250 MHz, CDCl₃): δ = 0.85 (d, 3H, J = 6.2 Hz, CH₃), 1.53–1.87 (m, 2H, CH₂), 1.96–2.17 (m, 2H, CH₂), 2.59–2.65 (m, 1H, CH), 4.62 (s, 1H, CH), 4.69 (s, 2H, NH₂), 6.92 (s, 1H, =CH), 7.25–7.43 (m, 8H, ArH); ¹³CNMR (62.9 MHz, CDCl₃): δ = 20.90, 28.81, 35.10, 35.55, 39.69, 59.24, 119.63, 120.19, 126.27, 126.71, 128.30, 128.59, 129.49, 129.77, 130.46, 130.68, 133.48, 134.08, 135.24, 135.55, 139.84, 140.89, 141.31, 159.36. Anal. Calcd For C₂₅H₂₅ClN₂O: C, 68.09; H, 4.76; N, 6.62; Found: C, 68.0; H, 4.78; N, 6.58.

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