INTRODUCTION

Glass is a ubiquitous material, used extensively in the architectural, automotive, aerospace, electronics, and chemical industries as a result of its transparency, esthetic appeal, chemical inertness, and interesting thermal and electrical properties. Flat glass is traditionally produced in sheet form via the float process whilst curved or customized shapes are cast in molds or are subtractively manufactured. The fabrication of bespoke molds, however, increases production costs while subtractive manufacturing can be time consuming and in both cases the geometries achievable cannot be easily customized. Additive manufacturing (AM) is useful in this regard as it facilitates the production of complex shapes without any prerequisite for molds and is cost effective when customization is important.1,2

Nevertheless, limited research has been conducted on AM of glass to date due to the challenges in glass processing. The high melting temperature of glass requires energy-intensive manufacturing procedures that create large temperature gradients within the built parts, generating residual stresses that can trigger thermal cracking. Additionally, the high transmittance in the visible and near-infrared range of the electromagnetic spectrum impedes the processing of glass with lasers operating in this range while any microcracks or porosity introduced within the built structures diminish their optical quality and lead to opaque surfaces.

The most promising AM methods for glass to-date employ a composite mix of silica particles to create a “green” part, which...
is thermally postprocessed to remove nonglass content. This approach has been successfully applied in stereolithography and direct ink writing, forming transparent and amorphous glass parts. However, postprocessing results in volumetric shrinkage, which leads to deviations between design and actual dimensions.

Processing of pure glass feedstock has also been investigated, utilizing a number of AM methods with varying levels of success. Deposition of molten glass through a motor-controlled nozzle resulted in transparent and esthetically pleasing parts intended for decorative applications. However, the increased layer thickness poses limitations in resolution for its use in other applications. Soda lime silica, borosilicate glass, and fused quartz walls have been produced with filament fed approaches; however, manual feeding of the filament under the laser beam restricted the process to elementary shapes with limited design freedom. Powder bed fusion methods have also been investigated for quartz, soda lime silica, and borosilicate glass, but these mostly concentrated on the formation of single layer scan tracks or basic multilayer forms.

This paper focuses on the development of a laser powder bed fusion approach (LPBF, also known as selective laser melting) for soda lime silica glass within a commercially available system. The main aim is to demonstrate the potential of the laser powder bed fusion of glass powder in forming complex multilayer glass structures. Section 2 below describes the feedstock material and the manufacturing and postprocessing methods while salient results and conclusions are presented in Section 3 and 4 respectively.

2 EXPERIMENTAL DESCRIPTION

A soda lime silica (72SiO₂−14Na₂O−10CaO−3MgO−1Al₂O₃) glass powder was used as feedstock material in this study. The powder is described by a Gaussian particle distribution with an average particle size of $D_{v,50} = 44 \, \mu m$ and bounds of $D_{v,10} = 29 \, \mu m$ to $D_{v,90} = 66 \, \mu m$ as determined by laser diffraction (Malvern, Mastersizer 3000, UK). This wide distribution denotes a good packing density; large amounts of fine particles reduce interparticle voids and provide a large surface area for laser processing. Additionally, the spherical particles of the powder (Figure 1) ensure adequate powder flowability. This was confirmed with a powder rheometer (Freeman technology, FT4 Powder Rheometer, USA), indicating a basic flowability energy (BFE) of 168 mJ and a marginally cohesive behavior, as indicated by a flowability rate index of (FRI) of 1.22, which can be attributed to the large amount of fine particles. UV-VIS and FT-IR tests identified an absorbance peak ($A = 10.2 \, \text{cm}^{-1}$) at 300 nm while an absorbance of $A = 0.4 \, \text{cm}^{-1}$ was found at the standard wavelength used in powder bed fusion (1064 nm) for bulk specimens of the same composition. Raw data for powder properties are available in the supplementary data file (Data S1).

The glass powder was processed in a commercial laser powder bed fusion system (ReaLizer GmbH SLM-50, Germany). The system uses a fiber laser operating at a wavelength of 1064 nm to selectively melt powder particles in layers of controlled thickness (glass transition temperature of 575°C and melting temperature of 1280°C were determined for the glass of this study by differential scanning calorimetry, Q600-TA Instruments, USA). A thickness of 70 μm was selected in this study to accommodate the largest particle size. The laser trajectories are defined by the slices of a stl model, made at 70 μm intervals through its thickness. At the end of each increment, the platform displaces vertically to allow an additional layer of powder to be deposited and the process is repeated to completion of the part.

Optimization of laser processing parameters (laser power, scan speed, and hatch spacing) was initially undertaken to determine the processing window for the formation of multilayer structures. The 2d and 3d energy density ED approaches (Equation 1a,b respectively) were used to populate process maps for thin walls ($7 \times 3 \times w \, \text{mm}^3$ where $w$: the width of a single laser scan track) and cubes ($5 \times 5 \times 5 \, \text{mm}^3$) respectively.

$$ED = \frac{P}{v \cdot t} \quad \text{and} \quad ED = \frac{P}{v \cdot h \cdot t}$$

where $P$: laser power (W), $v$: scan speed (mm/s), $h$: hatch spacing that is, the distance between adjacent laser scans (mm) and $t$: layer thickness (mm).
3 | RESULTS AND DISCUSSION

The process maps for the thin walls and cubes (Figure 2A,B respectively) apply to the soda lime silica glass of this study and are divided in regions depending on the quality of the produced parts.

For thin walls (Figure 2A), three regions are detected: (a) well-defined walls of uniform thickness and height; (b) partly consolidated walls that lack uniform height and; (c) combinations of processing parameters that could not consolidate the glass powder. Uniformity of wall height and thickness were respectively assessed qualitatively and quantitatively with optical microscopy; two specimens were analyzed per combination of parameters to ensure repeatability taking 10 equidistant readings per specimen for wall thickness. The thickness reported below is the average of the two specimens and is considered uniform when the coefficient of variation is below 10% for each specimen.

High-quality thin walls were found to be formed for energy densities exceeding $ED \geq 27 \text{ J/mm}^2$. The thin walls of this study were produced by repetition of a single laser scan in each layer; their thickness is, therefore, directly related to melt pool width, $w$, and this can provide insight on the resolution achievable for more complex structures produced through laser powder bed fusion. It was found that wall thickness was linearly proportional to the selected energy density and empirically follows Equation 2, while the minimum wall thickness that could be successfully achieved was 385 μm.

$$w = 0.00082 \cdot ED + 0.1597 \text{ (mm)}$$ (2)

Cubic structures were at first assessed qualitatively to define the energy density limits for successful material consolidation and distinguish well-defined parts: the lower limit was determined by parts that failed/cracked during the manufacturing process or their retrieval from the substrate while the upper limit was determined by parts that showed excessive track height/spheroidization in their center due to surface tension in the molten glass. Subsequently, quantitative analysis of the geometric accuracy followed up for the well-defined parts with Micro-CT imaging (ZEISS, XRAYA Versa XRM – 500, Germany) in slices of 4 μm thickness throughout the 5-mm height and image processing using the open source software ImageJ.15

It is found that cubic structures could be successfully produced for energy densities ranging between $80 \geq ED \geq 110 \text{ J/mm}^3$ (Figure 2B). Energy densities below this range resulted in brittle and poorly consolidated parts while energy densities above the maximum limit of this range led to excessive track height as previously described which impeded deposition of subsequent powder layers and thus, the completion of the AM process. The geometric accuracy of the average surface area of the Micro-CT slices (i.e., the difference between the design area and the actual area) was below 6% for all well-defined samples. The slight increase in surface area is attributed to unfused powder particles that adhere to the specimen.

Figure 2A,B provide useful information for manufacturing thin-walled or large surface area structures respectively, that have designs similar to the ones described above. Significant deviations from those for example, combinations of thin-walled and large surface area features within the same structure or structures with significantly large surface areas might require further fine-tuning of laser processing for example, multiple combinations of processing parameters within the same structure or optimization of laser scan strategies.

It should be noted here that the above process maps are developed for the soda lime silica glass powder of this study. Changes in particle size of the feedstock material will not change the required energy densities for material consolidation even when different combinations of parameters will have to be selected to accommodate the new layer thickness. However, different feedstock compositions and, therefore, material properties will lead to changes in the required energy density.

![Figure 2](https://example.com/figure2.png)
The next section demonstrates how the optimized processing parameters can be used to form complex structures that would not be achievable with standard glass processing methods. Lattice structures, designed from the previous trials using the in-house software Flatt_Pack, were successfully produced in different configurations. Examples of \(10 \times 10 \times 10\) mm\(^3\) BCC, gyroid matrix, gyroid network, and diamond network lattices are shown in Figure 3B. These structures can be also scaled up to suit different applications for example, the \(20 \times 20 \times 20\) mm\(^3\) gyroid network lattice shown in Figure 3C. Such structures are finding increasing academic interest and industrial application due to their unique properties for example, lightweight nature, high surface to volume ratio and energy-absorbing properties. Additionally, AM of lattice structures enables the control of cell geometry, grading, and conformance to external geometries not achievable with any other manufacturing method. However, to date these structures have been made from metal, polymer, or ceramic processing methods. This is the first demonstration that such structures can be generated from the selective laser melting of glass powder. The distinct properties of glass can open up new fields of application for these structures.

Complex AM structures have been previously demonstrated for ceramics. However, the main advantage of the LPBF method presented in this paper over AM methods commonly used for ceramics is the processing of pure feedstock material. Most AM methods for ceramics are indirect, involving binders to form a composite “green” part, followed by postmanufacturing heat treatment to remove binder content and sintering to densify the part. Pure ceramic feedstock has also been used in LPBF. However, parts of high complexity similar to the ones described in this paper have not been demonstrated.

Scanning electron microscopy (SEM) inspection demonstrated material consolidation in the built parts (Figure 3A) while partially fused powder particles could be observed on the lattice walls. X-ray diffraction (Bruker, D8 ADVANCE, USA) confirmed that the glass remained amorphous after laser processing (graph available in supplementary Data S1). The micro-CT slices were additionally analyzed in ImageJ to assess the internal porosity which marginally decreased with increased energy density; average porosities of 12.1% and 11.3% were respectively reported for energy densities of 78.5 and 107.1 J/mm\(^3\) which correspond to the energy density limits of the “well defined” region in Figure 2B.

The opaque appearance of the built parts can be jointly attributed to porosity and partially fused powder particles on the outer surface of the parts. Postprocessing of the built parts by heat treatment at 650°C for 5 hours can improve translucency whilst retaining part shape. However, this is accompanied by further material consolidation and, therefore, volumetric shrinkage.

The above findings can form the basis for tailoring laser powder bed fusion and postprocessing methods to suit particular applications for example, partially fused powder particles could act as baffles in micro-fluidic channels; an example of a fully enclosed channel produced with laser powder bed fusion are shown in Figure 3D with minimum permissible internal diameters of 1.1 mm. Additionally, highly porous glass structures with large surface to volume ratios could prove useful as structured catalysts in chemistry or scaffolds in tissue engineering, even in opaque forms, while smooth and translucent/transparent glass structures could be preferable in decorative applications. Tailoring of powder bed fusion method can be achieved by performing further investigations, similar to those described in this study on various glass compositions that would suit particular applications based on their properties.
4 | CONCLUSIONS

This paper successfully demonstrated the potential of a laser powder bed fusion approach to form glass structures with high levels of complexity in design that cannot be achieved with standard glass-forming methods. Process maps were populated for the formation of thin walls and solid cubes, which can be used as a guide for the formation of any small-scale glass structure. This study may open possibilities for a wide range of applications from chemistry and biomedical to decorative glass industries.

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SUPPORTING INFORMATION

Additional supporting information may be found online in the Supporting Information section at the end of the article.

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