Analysis of Constituents and Bacteriostatic Activity of Essential Oil Extracted from Sea Buckthorn Seeds by Different Methods

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Abstract: Sea buckthorn is a homologous plant listed in Chinese pharmacopoeia. The Essential Oils (EOs) of Sea Buckthorn Seeds (SBTS) are widely used in medicine and health products in China, Russia and other countries. The extraction methods of SBTS with advantages of low cost, high yield and high activity have attracted considerable attention. In this study, the extraction rate, chemical compositions and antibacterial activity of EOs from SBTS from Jilin Province extracted by Hydrodistillation (HD), Microwave-Assisted HD (MAHD) and Enzyme-Assisted HD (EAHD) were compared, so as to provide references for the preparation and application of EOs from SBTS. The results showed that the extraction rates of EOs from SBTS were (0.457±0.062%) when extracted by EAHD. The extraction rate of EAHD was significantly higher, with values reaching 2.3 and 1.4 times that of HD and MAHD, respectively. A total of 18, 26 and 26 volatile compounds were identified by Gas chromatography-mass spectrometer analysis. The three EOs were mainly composed of acid, ester and alkane and the acid component accounted for more than 50% of total, with the highest acid proportion occupied by HD (63%). The minimum inhibitory concentration of the EOs from SBTS extracted by HD, MAHD and EAHD against Bacillus subtilis, Bacillus pumilus and Staphylococcus aureus was determined simultaneously. The minimum inhibitory concentration of EOs from SBTS extracted by HD against Bacillus subtilis was 0.78 mg/mL, which is better than EOs extracted by MAHD and EAHD and consistent with the positive control. The minimum inhibitory concentration of EOs extracted by the three extraction methods to Bacillus pumilus was 1.56 mg/mL, which was lower than the positive control. The minimum inhibitory concentration of EOs from SBTS against Staphylococcus aureus was 3.12 mg/mL, which was lower than that of the positive control but better than that of EOs extracted by MAHD and EAHD. Based on the experimental results, the different extraction methods have a great influence on the extraction rate, chemical composition and activity of EOs. Although the extraction rate of EOs from SBTS extracted by HD was lower, it had low cost, more active components and better activity. Thus, these EOs can be developed as a natural antibacterial agent in the future.

Keywords: Sea Buckthorn Seeds, Essential Oils, Chemical Composition, Antimicrobial Activity

Introduction

Sea buckthorn (Hippophae rhamnoides L.) is an ancient plant with modern value and it is widely distributed in Asia and Europe. Given its unique nutrition and biological activity, Sea buckthorn has been used as a kind of traditional medicine in Russia and China (Marsiñach and Cuenca, 2019). Sea buckthorn, characterized by nitrogen fixation, cold resistance, salt and alkali resistance and wind and sand fixation, is a precious tree species for ecological forest construction (Fatima et al., 2012). Sea buckthorn fruit and Sea Buckthorn Seeds (SBTS) are rich in flavonoids, triterpenic acid, natural vitamins, Essential Oils (EOs), trace elements and other
functional active ingredients beneficial to the human body. Triterpenes have an anti-inflammatory activity (Chernenko et al., 2004). Flavonoids are found in many parts of sea buckthorn, with leaves having the highest flavonoid contents (Stobdan et al., 2013). The flavonoids in sea buckthorn have biological activities, including anti-aging, anti-fatigue, anti-tumor and immunity enhancement (Stobdan et al., 2013; Li et al., 2018). As a medicinal food homologous plant, sea buckthorn fruit is also widely used in fruit juice, jam and other food fields (Wang, 2015).

EOs are the main active ingredient in SBTS. The EOs of SBTS mainly contain acids, esters, alkanes, etc. Yue et al. (2017) extracted EOs from SBTS from Yijun, Shaanxi Province through Hydrodistillation (HD); the main components of EOs of SBTS are n-hexadecanoic acid, dibutyl phthalate, oleic acid and other unsaturated fatty acids and they exhibit a good antibacterial activity. Punia and Kumari (2017) summarized that the EOs of SBTS mainly contain fatty acids such as α-linolenic acid, palmitoleic acid and linoleic acid, which have antibacterial, antioxidiant and anti-atherosclerosis properties, increased activity of regenerative cell and other biological activities.

EOs are extracted through various way. Conventional HD, Microwave-Assisted HD (MAHD) and Enzyme-Assisted HD (EAHD) are common extraction methods for EOs. HD is the most commonly used means for extracting EOs from natural plants. Its process, equipment, operation and other technologies are relatively mature and have low cost and the equipment and operation are relatively simple (Jiao et al., 2012; Wang et al., 2010). MAHD works by radiating microwave- glitching solvents through the cell wall to the inside of the cell. As the solvent and cell fluid absorb the microwave energy, the cell wall rapture, freeing the active ingredients in the cell. MAHD has the advantages of short extraction time and high extraction rate (Wang et al., 2010; Golmakani and Rezaei, 2008). Enzymes can gradually degrade the cellulose-forming skeleton of the cell wall into glucose, destroy the skeleton structure of the cell wall and increase the dissolution of active components in cells. EAHD can decompose plant tissues gently, increase yield and extract active ingredients from plants to the maximum extent; thus, it has great application potential (Jiao et al., 2012).

To our knowledge, EOs from SBTS have been extensively studied, but no study reported about EOs from SBTS in Jilin. In addition, the preparation of EOs still presents challenges; HD exhibits disadvantages, such as long time, high energy consumption and low extraction rate (Della Porta et al., 1998; Chen et al., 2014). Related research has focused on the establishment of a high-efficiency, high-yield and high-activity extraction method for EOs from SBTS. Sea buckthorn in Jilin, China was used as the research object. The EOs of SBTS were extracted by HD, MAHD and EAHD. The EOs of SBTS were extracted by HD, MAHD and EAHD. The constituents of the EOs were analyzed by Gas Chromatography-Mass Spectrometer (GC-MS). The antimicrobial activities of EOs extracted by three different extraction methods were studied by using Bacillus subtilis, Bacillus pumilus and Staphylococcus aureus and the effects of these methods on the EOs of SBTS were investigated. The present study provides a theoretical basis for the development of sea buckthorn-related products in Jilin, China.

Materials and Methods

Materials and Reagents

SBTS were produced from a new variety Changbai Mountain No. 27 in Seabuckthorn Planting Base, Jiaohe City, Jilin Province, China and authenticated by Prof. Guangshu Wang, School of Pharmaceutical Sciences, Jilin University, Changchun, China. The samples were dried in a cool, ventilated container and the dried samples were crushed to powder.

Cellulase was purchased from Shanghai Yuanye Biotechnology Co. Dichloromethane was purchased from Tianjin Damao Chemical Reagent Factory. The strains, including Bacillus subtilis ATCC 6633, Bacillus pumilus ATCC 700814 and Staphylococcus aureus ATCC 29213 were purchased from Beijing Yuding Xinjie Technology Co. All reagents used were of analytical reagent grade.

Instruments

A GC-MS equipment (Shimadzu, Kyoto, Japan) and microwave synthesis reaction station at atmospheric pressure (Shanghai Xinyi Microwave Chemical Technology Co.) were used.

Culture of Strains

Bacillus subtilis, Bacillus pumilus and Staphylococcus aureus were obtained from Beijing Yuding Xinjie Technology Co. and cultured. A total of 20 µL microorganisms were absorbed by the micropipette and added to the corresponding medium. Before the bacteriostatic test, the three microorganisms were cultured for 24-48 h to ensure that they were active in their physiological period.

Preparation of EOs

Exactly 40 g SBTS powder and water were added to a 1000 mL round-bottom flask at a ratio of 1:15 and thoroughly mixed to prepare the extract. HD conditions: The above extracts were extracted by HD for 6 h; MAHD conditions: The above extracts were extracted with microwave power of 600 W for 60 min; EAHD conditions: The enzyme dosage was 0.77% SBTS and enzyme hydrolysis was conducted for 70 min at 55°C,
followed by extraction for 3 h. According to the general rules of four parts of Chinese Pharmacopoeia (2015 edition) 2204, the EOs were determined by the first method. After the extraction, dichloromethane was used to extract the oil layer and methylene chloride was distilled by a rotary evaporator to obtain the EOs of SBTS. The EOs were transferred to a brown bottle and sealed. The bottled was stored at 4°C for later use.

Equation (1) illustrates the formula used to calculate the extraction rate of EOs:

\[
\text{Extraction rate of } \text{EOs} = \left( \frac{m_1}{m_2} \right) \times 100\% \tag{1}
\]

Where:
- \(m_1\) = The weight of EOs extracted from SBTS
- \(m_2\) = The sample weight of SBTS

**GC-MS Analysis**

The contents of EOs from SBTS were studied by GC-MS. The following equipment and conditions were used: The chromatographic conditions involved the Rxi 5MS capillary column (30 m, 0.25 mm, 0.25 μm) with helium as carrier gas at 1 mL/min, sample volume 1.0 L. The GC oven temperature was kept at 6°C for 6 min, adjusted to 250°C at a rate of 3°C/min, then kept constant for 30 min. The injection port temperature was 250°C and the column pressure was 49.5 kPa. For MS, an electron ionization iSource at a temperature of 230°C was used. The electron energy was 70 eV and the scanning mass range was 40-500 m/z. The percent content of constituents was calculated automatically from peak areas of the total ion chromatogram. Constituents were identified from mass spectra and retention times using the NIST library.

The EOs extracted by the three methods were diluted 50 times with diethyl ether before the GC-MS analysis (Elkady and Ayoub, 2018).

Total Ion Current (TIC) was obtained by GC-MS in full scan mode. Each compound was determined by comparing the retention indices associated with the n-alkane mixture standard (C₈-C₄₀) and their mass spectra with reference data in the MS database (NIST 05) (Tian et al., 2019).

**Bacteriostatic Activity**

The antibacterial activity of EOs from SBTS extracted by three different methods against *Bacillus subtilis*, *Bacillus pumilus* and *Staphylococcus aureus* was determined.

The minimum inhibitory concentration, defined as the minimum concentration that inhibits the growth of visible microorganisms, was detected by broth microdilution assay. The EO of SBTS were tested on different bacteria. In a 96-well plate, 180 μL bacterial standard suspension (10⁸ CFU/mL) and 20 μL EO diluent were added to each well. In the first well, the concentration of EOs was 50 mg/mL. The concentration of EOs in the following 10 wells was reduced by serial two-fold dilution. The EO dissolved in 20% Dimethyl Sulfoxide (DMSO) and chloramphenicol (0.78-200 mg/mL) was the positive control and the 20% DMSO solution without EOs was the negative control. All assays were repeated thrice (Cui et al., 2018).

**Results**

**Extraction Rate of EOs**

Figure 1 shows the extraction rates of EOs extracted by HD, MAHD and EAHD. The extraction rates of EOs from SBTS obtained by HD (0.195±0.051%), MAHD (0.331±0.033%) and EAHD (0.457±0.062%) were determined. The extraction rate of EAHD was 2.3 and 1.4 times higher than that of HD and MAHD, respectively. The high yield rates of EAHD and MAHD may be due to the capability of enzymes to decompose plant tissues to maximize the flow of EOs (Sharma et al., 2002). Microwave radiation causes a rapid rise in the internal temperature of the cell; as a result, the cell wall ruptures and the EOs are released from the cell (Farhat et al., 2011).

**Analysis of the Compositions of EOs of SBTS Extracted by Three Different Extraction Methods**

Quantification was expressed as a percentage of the peak area. The relative content of each component (%) was calculated based on the peak area of GC. In terms of the TIC area, the EOs extracted by HD, MAHD and EAHD accounted for 91.78, 93.28 and 92.95% respectively. Table 1 shows the results.

![Fig.1: Oils yield of SBTS extracted by different methods. **P<0.01 compared with the HD](image-url)
Table 1: Composition of EOs extracted from SBTS by different extraction methods

| No. | Compound                                      | Molecular formula | Molecular weight | HD         | MAHD       | EAHD       |
|-----|------------------------------------------------|-------------------|------------------|------------|------------|------------|
| 1   | 9-Hexadecenoic acid                           | C₁₇H₃₅O₂           | 254              | 1.06       | 1.17       | 1          |
| 2   | Palmitic acid                                 | C₁₇H₃₆O₂           | 256              | 12.84      | 14.05      | 4.6        |
| 3   | Methyl 9-octadecynoate                        | C₁₉H₃₉O₂           | 294              | 2.64       | 1.9        |            |
| 4   | Hexadecanoic acid, butyl ester                | C₂₀H₄₀O₂           | 312              | 17.92      | 17.68      | 18.28      |
| 5   | Linoleic acid                                 | C₂₀H₄₁O₂           | 280              | 26.51      | 20.04      | 18.59      |
| 6   | Linolenic acid                                | C₂₀H₄₃O₂           | 278              | 3.68       | 4.7        | 4.85       |
| 7   | Stearic acid                                  | C₂₀H₄₂O₂           | 284              |            |            |            |
| 8   | Heneicosane                                   | C₂₁H₄₄             | 296              |            | 0.93       |            |
| 9   | 1-Heneicosyl formate                          | C₂₂H₄₆O₂           | 340              |            | 2.8        |            |
| 10  | Oleic acid                                    | C₁₈H₃₆O₂           | 282              |            |            | 1.24       |
| 11  | 4-Methyldecosane                              | C₂₀H₄₂             | 324              |            |            | 2.62       |
| 12  | 11,14-Eicosadienoic acid, methyl ester        | C₂₁H₄₂O₂           | 322              |            | 1          |            |
| 13  | 1-Heneicosanol,1-formate                     | C₂₂H₄₄O₂           | 340              |            | 1.22       |            |
| 14  | 5-Cyclohexylidocane                           | C₂₁H₃₉O₂           | 364              |            | 1.86       |            |
| 15  | Heptacosane                                    | C₂₃H₴₃             | 380              | 2.98       | 2.97       | 3.66       |
| 16  | 9,12-Octadecadienoic acid (9Z,12Z)-, 2,3-dihydroxypropyl ester | C₂₁H₃₈O₄           | 354              | 1.3        | 1.28       | 1.47       |
| 17  | Methyl (Z)-5,11,14,17-eicosatetraenoate       | C₂₁H₄₂O₂           | 318              |            |            | 4.93       |
| 18  | 9-(Z)-2-(12Z)-3-Hexenylidene] cyclopropylidene nonanoic acid 2-hydroxy-1-(hydroxymethyl)ethyl est | C₂₁H₄₂O₄           | 352              | 2.44       | 1.01       | 1.26       |
| 19  | Decyl octyl adipate                           | C₂₃H₄₄O₄           | 398              |            | 7.34       |            |
| 20  | 22-Tricosenoic acid                           | C₂₃H₄₄O₂           | 352              |            |            | 3.69       |
| 21  | (Z)-14-Tricosenyl formate                     | C₂₃H₄₄O₂           | 366              | 3.52       |            |            |
| 22  | Hexadecanoic acid, 1-decyl 6-octyl ester      | C₂₃H₄₂O₄           | 398              | 2.8        | 2.74       | 3.06       |
| 23  | 2,6-Octadiene, 2,6-dimethyl                  | C₁₀H₂₀             | 138              |            |            | 1.48       |
| 24  | 9-Eicosene-1,20-diol,1,20-diacetate            | C₂₀H₄₂O₄           | 396              | 2.76       |            |            |
| 25  | 1-Methyl-4-(1-methylethynyl) cyclohexanol     | C₁₀H₁₈              | 154              |            |            | 1.32       |
| 26  | 2,2'-Methylenebis(6-tert-buty1-4-methylphenol) | C₂₃H₴₂O₂           | 340              | 2.45       | 0.9        | 4.27       |
| 27  | Stigmastane-3,6-dione,(5a)-                   | C₂₃H₄₂O₂           | 428              |            |            | 1.12       |
| 28  | Cycloketocrasone                              | C₂₃H₴₂O₂           | 336              |            |            | 2.1        |
| 29  | Dotriacontane                                 | C₂₃H₴₂O₂           | 450              | 1.31       | 1.07       | 2.19       |
| 30  | 1-bromotriacontane                            | C₂₃H₄₂Br           | 501              |            |            | 1.08       |
| 31  | di-n-Decyl phthalate                          | C₂₃H₄₂O₂           | 446              |            |            | 1.53       |
| 32  | Tetraiacontane                                | C₇₃H₄₂O₂           | 478              |            |            | 1.05       |
| 33  | Pentriacontane                                | C₂₃H₄₂O₂           | 492              |            |            | 1.56       |
| 34  | 2,5-Furandione,3-(2-dodecen-1-yl)dihydroxy-   | C₁₄H₂₆O₃           | 266              |            |            | 1.85       |
| 35  | 4-Methyl-4-hydroxycyclohexanone               | C₁₄H₂₆O₃           | 226              |            |            | 1.23       |
| 36  | 9-Octadecenoic acid (9Z)-, tetradecyl ester   | C₂₂H₄₂O₂           | 478              |            |            | 2.08       |
| 37  | Hexatriacontane                               | C₂₃H₄₂O₂           | 506              | 1.14       | 1.28       |            |
| 38  | 1,2-Benzenediacarboxylic acid, ditericdeyl ester | C₂₂H₄₂O₂           | 530              |            |            | 1.2        |
| 39  | 17-Pentatriacontone                           | C₂₃H₄₂O₂           | 490              | 1.88       |            | 2.48       |
| 40  | Tetracontane                                  | C₁₀H₂₈             | 563              | 1.67       | 2.71       | 1.67       |
| 41  | Dotriacontane                                 | C₁₀H₂₈             | 591              |            | 0.9        |            |
| 42  | Trietacontane                                 | C₁₀H₂₈             | 618              |            |            | 1.29       |
| 43  | Hexadecanoic acid, 1,1’-(2-hydroxy- 1,3-propanediyl) ester | C₁₅H₃₄O₃           | 568              |            |            | 1.57       |
| 44  | 1-Hentetracontanol                            | C₁₀H₂₈O₂           | 593              |            |            | 0.88       |

Table 2: Composition of EOs extracted from SBTS by different extraction methods

| Acids (%) | Esters (%) | Alkane (%) | Alcohols (%) | Others (%) | Total (%) |
|-----------|------------|------------|--------------|------------|-----------|
| HD        | 65.81      | 6.02       | 8.98         | -          | 10.97     | 91.78     |
| MAHD      | 57.56      | 11.35      | 14.84        | 0.88       | 8.57      | 93.28     |
| EAHD      | 54.99      | 10.23      | 18.06        | 1.32       | 8.35      | 92.95     |

As presented in Table 2, the EOs extracted from SBTS by HD, EMHD and MAHD were mainly composed of fatty acids, esters and alkanes, among which fatty acids are the most common, with the contents of 65.81% (HD), 57.56% (MAHD) and 54.99% (EMHD), respectively. Compared with the findings of
Yue et al., the content of fatty acids in the EOs extracted from SBTS increased, which might be caused by different species and growing environment of sea buckthorn. Wen et al. (2018) compared the chemical constituents of the EOs of Saussurea sinensis from different production areas and observed that due to the influence of temperature, sunshine, latitude, altitude, soil, vegetation, precipitation distribution and cultivation technology level in different regions, the chemical constituents of EOs differed to a certain extent. The fatty acids extracted by the three methods were the same and mainly included linoleic acid (17.92, 17.68 and 18.28%), linolenic acid (26.51, 20.04 and 18.59%), palmitic acid (12.84, 14.05 and 4.06%). The extraction mechanism of different extraction methods may be the reason for the different fatty acid contents. Xiang et al. (2010) analyzed the EOs of propolis by HD and supercritical carbon dioxide extraction and the results showed significant differences in the EOs components obtained by these extraction methods. Thus, in industrial production, we should not only pay attention to the extraction rate but also the importance of component contents.

Fatty acids have various physiological functions. Linoleic and linolenic acids are essential fatty acids that cannot be synthesized by themselves and can only be obtained through diet (Marsiñach and Cuenca, 2019). Linolenic acid is a physiological component of the cell membrane and mitochondrial membrane and is involved in cell transport and neuronal signal transmission (Zielińska and Nowak, 2014). This compound has an antihypertensive effect on blood pressure. Linoleic acid is the most abundant polyunsaturated fatty acid in human skin. A lack of linoleic acid in diet can lead to typical squamous skin disease and excessive epidermal water loss (Ziboh et al., 2000). Linolenic and linolenic acids play an important role in bone development (Griel et al., 2007; Lavado-García et al., 2018).

Antibacterial Activity of EOs from SBTS Extracted by Three Different Extraction Methods

As shown in Table 3, the minimum inhibitory concentration of EOs from SBTS extracted by HD against Bacillus subtilis was 0.78 mg/mL, which was better than and half that of the minimum inhibitory concentration of EOs from SBTS extracted by MAHD and EAHD. This result is also consistent with the positive control. The minimum inhibitory concentration of EOs from SBTS extracted by the three methods to Bacillus pumilus was 1.56 mg/mL, which was less than that of the positive control and twice of its minimum inhibitory concentration. The minimum inhibitory concentration of EOs from SBTS extracted by HD against Staphylococcus aureus was 3.12 mg/mL, which was lower than that of the positive control but better than that of EOs extracted by MAHD and EAHD.

The antibacterial effect of EOs from SBTS may be related to their rich free fatty acids, several of which can kill or inhibit the growth of bacteria. The main targets of free fatty acids are cell membranes, where they disrupt electron transport chains and oxidative phosphorylation. In addition to interfering with cell energy production, free fatty acids may inhibit enzyme activity, damage nutrient absorption, produce peroxidation and autooxidation degradation products, or directly lyse bacterial cells (Desbois and Smith, 2010). Linoleic and linolenic acids have good bacteriostatic effects on certain strains (Amgalaanbaatar et al., 2014; Jung et al., 2015; Jang et al., 2016).

### Table 3: The minimum inhibitory concentration and positive control of EOs from SBTS against different bacteria (mean value, mg/mL)

|            | Bacillus subtilis | Bacillus pumilus | Staphylococcus aureus |
|------------|------------------|------------------|----------------------|
| HD         | 0.78             | 1.56             | 3.12                 |
| MAHD       | 1.56             | 1.56             | 6.25                 |
| EAHD       | 1.56             | 1.56             | 6.25                 |
| Chloramphenicol | 0.78           | 0.78             | 0.78                 |

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**Conclusion**

The EOs from SBTS were extracted by HD, MAHD and EAHD. The effects of these different extraction methods on the extraction rate, composition and antibacterial activity of the EOs from SBTS were studied. The results showed that the extraction yield of EAHD was the highest and the extraction yields of MAHD and EAHD were significantly higher than that of HD. A total of 18, 26 and 26 chemical components were identified from the EOs extracted by HD, MAHD and EAHD, respectively. The three extracts all contained acids, esters, alkanes and other compounds, but the contents of components were significantly different. The EOs extracted by the three methods had antibacterial effect on Bacillus subtilis, Bacillus pumilus and Staphylococcus aureus. The antibacterial effect of the EOs extracted by microwave and enzyme-assisted extraction was lower than that of the positive control. Compared with those obtained by the other methods, the EOs extracted by HD had a better inhibitory effect on Bacillus subtilis, consistent with the positive control. The antibacterial effect of EOs extracted by HD on Staphylococcus aureus was better than that of the EOs extracted by the other two methods but is inferior to that of the positive control. Bacillus subtilis, Bacillus pumilus and Staphylococcus aureus are representative Gram-positive bacteria, especially Staphylococcus aureus. Staphylococcus aureus is one of the most important pathogenic bacteria in humans and it can cause numerous serious infections (Tomlinson et al., 2012;
Grundmann et al., (2006). EOs extracted from SBTS have antibacterial effect on *Staphylococcus aureus*, which indicates that they have potential to be developed as nature-based antibacterial agents. In conclusion, different extraction areas and extraction methods have great effects on the extraction rate, components and antibacterial activity of volatile oils from SBTS. Despite the assumption that high yield and good activity are consistent, the experiment showed that the extraction yield of SBTS oil obtained by the three methods is not necessarily related to antimicrobial activity. In the future, when the EOs from SBTS are developed and utilized in the fields of medicine and food, the extraction method can be determined based on the level analysis of the target composition and extraction rate to improve the EOs of SBTS as antibacterial agents, promote epithelial tissue regeneration, burn and reduce blood lipids and exert an anti-tumor effect (Du et al., 2004; Korneyev, 2003).

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**Author’s Contributions**

Hepeng Zhao and Fuyin Zhao: Extracted and analyzed the constituents of EOs from SBTS and studied their antibacterial activity.

Qian Xiao: Revised and proofread the thesis.

Hongli Zhou and Xiudong Yang: Guided the design route and provided experimental guidance for this manuscript.

**Ethics**

This article is original and contains unpublished material. The corresponding author confirms that all of the other authors have read and approved the manuscript and no ethical issues involved.

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