Original paper

First Report of using Response Surface Methodology for the Biodegradation of Single Azo Disperse Dyes by Indigenous Daedalea dickinsii-IEBL-2

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Abstract

Textile dyes accounts for 22 % of total dyes utilization and are potential pollutants due to their toxicity and carcinogenicity. Brown Rot Fungi (BRF) have great potential of biodegradation of dyestuffs due to their efficient nonspecific ligninolytic enzyme system. Current study was conducted for application of indigenous Daedalea dickinsii IEBL-2, a brown rot fungi, to degrade single azo disperse dyes i.e. disperse violet-63 and disperse orange-30. Response Surface Methodology (RSM) with Box Behnken Design was employed to optimized biodegradation process. Ligninolytic enzymes involved in biodegradation were also studied and biodegradation process was monitored by High Performance Liquid Chromatography (HPLC). There was 72.14±1.1% and 68.71±1.04% biodegradation observed after screening experiments for dye-1 and dye-2. Optimization by RSM at step-1 increases biodegradation to 80.83±1.35% and 78.3±1.16% and after step-2, 88.93±1.32% and 93.32±1.54% respectively. Study of ligninolytic enzymes i.e manganese peroxidase, lignin peroxidase and laccase revealed their involvement in biodegradation process. HPLC analysis confirmed the biodegradation process and conversion of complex dyes into simple secondary amines i.e diphenylamine and 3-methyldiphenylamine. These findings would help to reduce the environmental pollution by textile effluent and also reduces the risk of certain diseases.

Keywords Textile dyes, response surface methodology, ligninolytic enzymes, Daedalea dickinsii, biodegradation.

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Introduction

Environmental pollution creates many problems including different threatening diseases. Reduction in environmental pollution is a great challenge for human beings and it is the main focus of the scientists. Textile industrial effluent is one of the major causes of environmental pollution and textile dyes are more difficult to treat due to their stability being synthetic [1-3]. Water is one of the major necessities of life for plants, animals and human beings. Exposure of human body to polluted water for a long period of time, lead to certain diseases like cancer. Conventional wastewater treatment processes are often ineffective in dyes removal from the effluent [1]. Conventional techniques like, ozonation, electrolysis and coagulation are often costly, limited applicability and produce more toxic secondary amines [4, 5]. Bioremediation is becoming important, because it is cost-effective and environmentally friendly and produces less sludge [1]. Many microorganisms are capable of decolorizing the azo dyes, including gram-positive bacteria, gram-negative bacteria and fungi. The possibility to use ligninolytic fungi for the removal of synthetic dyes is one approach that attracts considerable attention [6].

Brown rot fungi are efficient in biodegradation of recalcitrant compounds like xenobiotics, lignin and dye stuffs by their extra cellular ligninolytic enzyme system [7]. The objectives of the current study was utilization of Daedalea dickinsii IEBL-2 for the biodegradation of disperse dyes. Optimization of biodegradation and study of effect of different parameters by using RSM. Study of ligninolytic enzymes secreted by fungi during biodegradation and their characterization to make them more applicable for industrial processes. Conformation of biodegradation of complex dyes by HPLC of treated and untreated sample.

Materials and Methods

Source of Daedalea dickinsii IEBL-2
Daedalea dickinsii-IEBL-2 was obtained from Industrial and Environment Biotechnology Laboratory (IEBL), Department of Biochemistry, PMAS-Arid Agriculture University Rawalpindi. Previously it was collected from Islamabad Pakistan, cultured on Malt Extract Agar media, identified morphologically from Department of Plant Pathology, PMAS-AAUR and preserved in IEB Laboratory for future use [8].

Disperse dyes for biodegradation experiments
Single azo disperse dyes i.e. disperse violet-63 (Violet S.RL) and disperse orange-30 (Yellow brown S.RFL) were purchased from supplier and used for biodegradation experiments. These dyes have molecular weight 414.85 and 450.27 while λmax 448 nm and 612 nm respectively.

Screening of biodegradation of disperse dyes
Biodegradation potential of D. dickinsii IEBL-2 for both single azo disperse dyes, the screening experiment were performed for 10 consecutive days at pre-optimized conditions (pH 5.5, temperature 28°C, dye conc. 0.02%, fungal inoculum 2 mL).

Optimization of biodegradation
Response Surface Methodology (RSM) was used for the optimization of various parameters as well as biodegradation process. RSM is a collection statistical and mathematical tools that use to designed experiments where there is one dependent variable and many independent variables [8]. The experimental design through RSM reduces the number of treatments and time for complete optimization of all parameters [9]. Box Bhenken Design (BBD) was used to optimize 5 parameters simultaneously and studies their effect on biodegradation [10]. Following parameters were optimized and their effect was monitored to enhance the biodegradation of disperse dyes.

Optimization of growth and nutritional parameters
Growth of D. dickinsii IEBL-2 dependent on different parameters like pH of media [11], temperature and size of inoculum [12]. Two additional carbon sources glucose and fructose were optimized along with three nitrogen sources including ammonium sulfate, ammonium nitrate and ammonia [13]. Various concentrations (0.01%-0.05%) were added in biodegradation mixture to monitor its effect on biodegradation [14].

Study of ligninolytic enzymes
To study the enzymes involved in biodegradation of disperse dyes, ligninolytic enzymes including lignin peroxidase (LiP), manganese peroxidase (MnP) and laccase were studied. Lignin peroxidase activity was calculated by method of Tein and Kirk 1984 [15] while manganese peroxidase activity was calculated according to method as described by WARSHII et al, 1992 [16]. Laccase activity was measured according to procedure of WOLFEDEN and WILSON, 1982 [17].

Purification of enzymes
Ligninolytic enzymes were partially purified by ammonium sulfate precipitation and gel filtration chromatography. Varying concentrations of ammonium sulfate (20-70%) were added in crude extract and place for overnight and pellet was used for enzymes assays [18]. Gel filtration chromatography was performed with sephadex G-100 column packed with sodium phosphate buffer, elutions collected after passing 2 ml of crude extract and subjected to enzymatic assays [19].

Characterization of enzymes
Partially purified LiP, MnP and Laccase were characterized for the most suitable pH of each enzyme and the most suitable temperature. For optimum pH enzymatic activities were performed from pH 4-6.5 and for optimum temperature 20°C-40°C [2]. Effect of varying concentration of substrate (2 mM to 10 mM) was also monitored to find Km and Vmax of each of the enzymes. Well characterized enzymes are more suitable for different industrial applications [19].

HPLC analysis of biodegraded dyestuff
Textile dyes usually convert into secondary amines after biodegradation by lignolytic enzymes. To confirm the biodegradation of dyes under study, biodegraded material was subjected to HPLC analysis before and after treatment.

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The peaks obtained were compared with peaks of available standards of secondary amines [20]. The HPLC was performed on SHIMAZDU LC-20AT model having C-18 column of 25 mm length with internal diameter of 4.6 mm. The analysis was performed in isocratic and reverse phase mode at 1 mL/min flow rate. The solvent system was acetonitrile: water (60:40) ratio and 20 μL of each sample was injected for analysis.

Results and Discussion

Screening of biodegradation potential of *D. dickinsii* IEBL-2

Both dyes (Fig. 1a & 1b) were screened for the biodegradation potential by *D. dickinsii* IEBL-2. The screening conditions for each experiment were 90 mL dye solution (0.02%), 10 mL of liquid media, 2 ml fungal inoculum, pH 5.5, temperature 28°C and 150 rpm [2, 8]. There was 72.14±1.1% biodegradation achieved for D-1 and 68.71±1.04% for D-2 after 10 days (Fig. 2). The increase is prominent up to day 7 than very low rise in biodegradation possibly due to accumulation of toxic waste and depletion of media [2, 21]. During screening LiP observed the most active enzymes with maximum activity 865.5 U/mL/min followed by MnP (612. 31 U/mL/min) and laccase (312.71 U/mL/min) (Fig. 3). Previous studies also reported the role of these enzyme in textile dyes biodegradation. Decrease in degradation after day 7 might be related with decrease in production of enzymes [22].

![Figure 1. Structure of single azo disperse dyes (a) disperse violet 63 (b) disperse orange 30.](image)

![Figure 2. Screening of biodegradation of single azo disperse dyes by *D. dickinsii* IEBL-2.](image)

![Figure 3. Activities of lignolytic enzymes produced by *D. dickinsii* IEBL-2 during screening experiments.](image)

Optimization of biodegradation of disperse violet 63

There was 9% increased in biodegradation observed after optimizing pH, temperature, time period, inoculum size and dye conc. through RSM. Response surface graphs (3D) showed that different parameters interact positively with each other and increased biodegradation by enhancing fungal growth. Considering the interaction between pH of biodegradation mixture and size of fungal inoculum, there was maximum biodegradation i.e. 78.67% at pH 5.2 and 6.89 mL (Fig. 4a). The side lines indicate the individual effect of change in pH and change in inoculum size on biodegradation. Up rise in graph indicate the positive interaction between these parameters which increase the biodegradation. If consider pH and temperature for their effect on biodegradation, the maximum biodegradation (79.49%) observed at pH 5.36 and 28.75°C with positive interaction (Fig. 4b). Similarly, very prominent interaction was observed between time period (days) and inoculum size, at 7.2 mL and 7.8 days there was 80.12% biodegradation observed (Fig. 4c). Same positive effect was present between dye conc. and inoculum (Fig. 4d) and also between dye conc. and temperature (Fig. 4e). When all 5 parameters collectively, maximum biodegradation 80.83±1.35% obtained at pH 5.25, inoculum size 7.12 mL, time period 8 days, temperature 26.59°C and dye conc. 0.046%. There was significant effect of these parameters on biodegradation indicated by high F. ratio (22.26) and P <0.0001 at R²=95.23% (Fig. 4f). There must be selection of right values of these parameters to achieve maximum biodegradation [23, 24].
Effects of five carbon and nitrogen sources were check simultaneously. There was 8% increased in biodegradation of disperse violet 63 dye by *D. dickinsii* IEBL-2. Various parameters interact positively and nearly 1% concentration of each of glucose, fructose, ammonium sulfate, ammonium nitrate and ammonia were found the most suitable for biodegradation (Fig. 5). Furthermore, concentration of ammonia more than 1% has negative effect possibly due to denaturing effect on proteins. There are many studies report the positive impact of above mentioned sources on growth of fungi and biodegradation of textile dyes by these fungi [24, 25].
Optimization of biodegradation of disperse orange 30

The biodegradation of disperse dye-2 was also optimized by treating with *D. dickinsii* IEBL-2 and there was 10% increase in biodegradation achieved after optimization (78.3±1.16%) comparing with screening results (68.71±1.04%). The study of 3D response surface graphs showed that there was positive interaction between different parameters which lead to maximum biodegradation (Fig. 6). The maximum achieved biodegradation % age was the combined effect of these parameters rather than individual affect [2, 5, 19]. The most suitable conditions observed for maximum biodegradation were pH 5.32, inoculum size 6.78 mL, time period 8 days, temperature 27.11°C and dye conc. 0.04%. The statistical analysis of the results indicated high F ratio (21.43), very low P value (P<0.0001) and high $R^2$ ($R^2 = 94.11$) revealed that parameters under studies significantly affect the biodegradation process [25, 26].

In current study there was significance increased (15%) in the biodegradation of disperse orange 30 after addition of suitable conc. of different carbon and nitrogen sources (Fig. 7). Like disperse dye-1, nearly 1% amount of each of source gave maximum biodegradation i.e. 93.32±1.54%. Higher concentration of nitrogen sources negatively affect the biodegradation process due to non-stability of enzymes responsible for biodegradation.

**Study of ligninolytic enzymes during optimization of biodegradation**

The biodegradation of disperse violet 63 and disperse orange 30 depends upon the activity of ligninolytic enzymes which was confirmed by increasing the activity of enzymes with biodegradation [27]. Fungus produces three ligninolytic enzymes which degrade the complex structure of dye by their combined activity. These enzymes are non-specific and has the ability to oxidase complex aromatic compounds like textile dye and render them non-toxic. Maximum ligninolytic activities observed after optimizing of various growth factors with RSM were 955.27±6.21 UmL⁻¹min⁻¹, 721.15 ±3.22 UmL⁻¹min⁻¹ and 391.45±3.26 UmL⁻¹min⁻¹ for LiP, MnP and laccase respectively (Table 1). The increased observed after optimizing carbon and nitrogen sources was 78 UmL⁻¹min⁻¹, 47 UmL⁻¹min⁻¹ and 87 UmL⁻¹min⁻¹ for LiP, MnP and Laccase (Table 1). These results indicated that all three ligninolytic enzymes actively participated in the biodegradation of aromatic dyes present in the effluents. High activity of LiP compared to other two ligninolytic enzymes indicated its more roles in biodegradation [28, 29].

**Figure 6.** Response surface 3D graphs showing interaction between various parameters during the biodegradation of disperse orange 30 dye (D-2) by *Dadaelea dickinsii* IEBL-2, graphs represents positive interaction between (a) inoculum size and pH (b) days and pH (c) days and inoculum size (d) inoculum size and dye conc. (e)- day and temperature (f)- statistical analysis of results showing significant effect of parameters on biodegradation.

| Source     | DF  | Sum of Squares | Mean Square | F Ratio | Prob > F |
|------------|-----|----------------|-------------|---------|----------|
| Model      | 20  | 18481.615      | 924.081     | 21.435  | <0.0001  |
| Error      | 25  | 1977.846       | 47.114      | 0.116   |          |
| C. Total   | 45  | 19559.460      |             |         | <0.0001  |

$R^2 = 94.11\%$
Figure 7. Response surface 3D graphs showing interaction between carbon and nitrogen sources during the biodegradation of disperse orange 30 dye (D-2) by Dadaelea dickinsii IEBL-2, graphs represents positive interaction between (a) glucose and fructose (b) glucose and ammonium sulfate (c) fructose and ammonium nitrate (d) fructose and ammonia (e)- ammonium sulfate and ammonium nitrate (f)- statistical analysis of results showing significant effect of parameters on biodegradation.

### Table 1. Summary of ligninolytic activities during biodegradation process and purification

| Activities of ligninolytic enzymes by D. dickinsii IEBL-2 (IU/mL/min) | Step 1       | Step 2       | Step 3       | Step 4       |
|---------------------------------------------------------------|--------------|--------------|--------------|--------------|
| LiP                                                           | 955.27 ±6.21 | 1053.9 ±4.32 | 1257.3 ±4.53 | 1487.4 ±3.98 |
| MnP                                                           | 721.15 ±3.22 | 768.43 ±3.78 | 989.65 ±3.98 | 1232.2 ±3.87 |
| Laccase                                                       | 391.45 ±3.26 | 478.8 ±4.09  | 622.21 ±3.01 | 789.43 ±4.32 |

Where:
- Step 1: After optimization of initial conditions
- Step 2: After optimization of carbon and nitrogen sources
- Step 3: After ammonium sulfate precipitation
- Step 4: After gel filtration chromatography

**Purification of ligninolytic enzymes**

Ligninolytic enzymes were partially purified by ammonium sulfate precipitation and gel filtration chromatography in order to enhance their activities and made them more effective for biodegradation. Ammonium sulfate was added in the mixture in different percentage from 20% to 70% and activity assay was performed after overnight incubation with pellet. Addition of 60% ammonium sulfate gave maximum purification of ligninolytic enzymes from *D. dickinsii* IEBL-2 due to which activities of ligninolytic enzymes (Table 1) increased compared to activities after optimization of biodegradation [30].

Partially purified ligninolytic enzymes from *D. dickinsii* IEBL-2 were subjected to gel filtration chromatography having column packed with Sephadex G-100. Elutions obtained were used to performed activity assay to monitored purification process. Maximum activities were obtained in elution 12 (LiP), 14 (MnP) and 17 (laccase), as per size of different enzymes (Table 1) [8].

**Characterization of ligninolytic enzymes**

Ligninolytic enzymes were characterized to find their optimum pH, temperature and kinetic values (Km and Vmax). The results showed that pH 5.5 is the most suitable for all three enzymes i.e. LiP, MnP and laccase (Fig. 8). The LiP was found most active enzyme with higher enzymatic activity as compared to other two enzymes. These enzymes showed good activity between pH 4 to 6.5, after that activity became very low [31]. It was observed that Lip, MnP and laccase gave maximum activities at 28°C temperature maintaining during activity assay (Fig. 9). The activity of remain fairly high from temperature 26°C to 32°C and there was decreased in activity after that [31].
The kinetic studies of Lip showed that it has \( K_m = 0.751 \) mM and \( V_{max} = 1250 \) \( \mu \)M/mL/min while, MnP has \( K_m = 0.700 \) mM and \( V_{max} = 1000 \) \( \mu \)M/mL/min (Fig. 10). The kinetic studies of laccase showed that it has \( K_m = 0.571 \) mM and \( V_{max} = 1428.57 \) \( \mu \)M/mL/min (Fig. 10).

Lowered values of \( K_m \) indicated the higher affinity of ligninolytic enzymes towards their substrates. Higher values of \( V_{max} \) indicated the higher rates of enzymes catalyzed reactions under optimized conditions [31, 32].

**Figure 8.** Optimization of pH for the lignolytic enzymes produced by *D. dickinsii* IEBL-2.

**Figure 9.** Optimization of temperature for the lignolytic enzymes produced by brown rot fungi during biodegradation process.

**Figure 10.** Line-weaver Burk Double Reciprocal plot between \( 1/V_0 \) and \( 1/[S] \) for (a) Lignin peroxidase (b) Manganese peroxidase and (c) Laccase to determine the values of kinetic parameters i.e. \( K_m \) and \( V_{max} \).

**HPLC analysis of dyestuff before and after biodegradation**

The biodegradation process can be monitored by HPLC analysis of untreated and treated samples of textile dyes. Biodegradation of disperse textile dyes usually generated secondary amines, which may be less toxic than dyes or non-toxic. The study of secondary amine as standards with HPLC and comparison of graphs with treated samples would help to detect compounds generated after biodegradation [33]. Three available secondary amines were used as standards i.e. N-Methylaniline, 3-Methylphenylamine and Phenylamine. The retention time of N-Methylaniline was \( t_R = 3.57 \) min, that of 3-Methylphenylamine was \( t_R = 3.87 \) min and Diphenylamine was \( t_R = 3.69 \) min (Fig. 11). Retention time of each compound depends upon its affinity towards column as well as the mobile phase used for elution [27].
The retention time of untreated disperse dye-1 was $t_R=3.33$ min while its treated sample with *D. dickinsii* IEBL-2 gave two known compounds i.e. 3-methyldiphenylamine and diphenylamine as well as several small peaks of unknown compounds (Fig. 12). The retention time of untreated disperse dye-2 was $t_R=3.29$ min and treated dye-2 with *D. dickinsii* IEBL-2 results into diphenylamine and one unknown compounds (Fig. 13).

**Figure 11.** HPLC chromatogram of standard secondary amines (a)- 3-Methyldiphenylamine ($t_R=3.87$ min) and (b)- Diphenylamine ($t_R=3.69$ min).

**Figure 12.** HPLC chromatogram of disperse dye-1 (a)- Untreated sample (b)- treated sample with *D. dickinsii* IEBL-2, peak 2 for diphenylamine ($t_R=3.69$ min), peak 3 for 3-methyldiphenylamine ($t_R=3.69$ min) and peaks 4-7 are unknown compounds.
Figure 13. HPLC chromatogram of disperse dye-3 (a)- untreated sample (b)- treated sample with D. dickinsii IEBL-2, peak 1 for diphenylamine ($t_R=3.69$ min) and peak 2 ($t_R=4.72$ min) for unknown compound.

The secondary amines produced during the biodegradation of disperse textile dyes under study are less toxic. 3-Methyl diphenylamine may cause small irritation during long exposure but usually it is considered as non-carcinogenic and non-mutagenic. While Diphenylamine reported as non-carcinogenic and non-mutagenic but cause irritation or toxicity during long exposure to organs [34, 35].

Conclusions

Brown rot fungi is suitable microorganisms with ability of biodegradation of toxic textile dyes in eco-friendly way. D. dickinsii IEBL-2 has the potential to degrade disperse dyes more than 85% by using its ligninolytic enzymes. HPLC analysis confirms the biodegradation and production of less toxic by-products.

Conflict of Interest

Authors declares no conflict of interest.

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