FTIR-spectrophotometric analysis of levosulpiride and its pharmaceutical formulations

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Abstract
A fourier transform infrared (FTIR) spectrometric method was developed for the rapid, direct determination of levosulpiride in different pharmaceutical products.¹³ The universal ATR spectra were recorded and used for this study. Multiple linear regression (MLR) where a restricted set of absorption band is used for calibration beer-lambert law was used for data processing. A recovery of 98.5% of levosulpiride from the tablet dosage form with a correlation coefficient of 0.996 was obtained. The aim of this study was to develop a FTIR spectrophotometric procedure for the analysis of levosulpiride in raw material and tablets. The linear regression equation for levosulpiride was calculated to be y=189.044+152.39*x where x and y are concentration and integrated peak area respectively. The method had excellent reproducibility for the standard of 50 mg, 49.5 ±1.02 mg (n =6). The recovery test is an experimental design to verify the relationship between the amount of substance added and the amount quantified by this assay. In this test the observed concentrations of pure levosulpiride in the powdered tablets were not significantly different from the stated concentrations by student’s t test, p = 0.05% (100.04 ± 1.49%, n = 3) the method gave rise to linear data in the range 2-50. Mg with accuracy and precision in the range 0.77-3.2%. Therefore, this FTIR-spectrophotometric assay was accurate, and may be recommended for the simple quantification of levosulpiride.

Keywords: FTIR, Levosulpiride, ZNSE, AUC.

Introduction
Levosulpiride is chemically N-[(1-Ethylpyrrolidin-2-yl) methyl]-2-methoxy-5-sulfamoylbenzamide¹ (Fig. 1). It is a substituted benzamide antipsychotic reported to be a selective antagonist of dopamine D₂ receptor with activity on both central and peripheral levels.²

It is an atypical neuroleptic and a prokinetic agent.³ Levosulpiride is also claimed to have mood elevating properties at low doses.⁴ Levosulpiride increases dopaminergic neurotransmission, primarily by the blocking of the dopamine autoreceptors, which inhibits the pre-synaptic dopamine synthesis and release of dopamine. Compared with racemic and dextro-forms, the levo-form of sulphiride has greater central anti-dopaminergic activity,⁵ antiemetic, dyspeptic effects and lower acute toxicity.

Literature survey reveals that various method for analysis of levosulpiride including HPTLC,⁶ HPLC⁷ and UV spectrophotometetr.⁸,⁹ But this method proves to be more complex and is uneconomical. Here we propose an additional method for the quantification of Levosulpiride in bulk as well as formulation by Fourier Transform Infrared spectrophotometer using Zinc - Selenide crystal ATR method, which is simpler and produce accurate results.

Infrared spectrometry (IR) provides an useful way for the identification of drugs. However, the traditional techniques employed to obtain the IR spectra, such as alkali halides disks, mulls and thin films, are sometimes not adequate for quantitative analysis.¹⁰ Fourier Transform (FTIR) permits continuous monitoring of the spectral baseline and simultaneous analysis of different components of the same sample.¹¹,¹²

FTIR spectrometry provides information about components present in formulation. Chemometric methods, such as principal component regression (PCR), Improved Principal Component Regression and partial least squares (PLS2, Multicomponent Partial Least Squares) analysis are commonly used to extract the specific information relevant to the analyte of interest from the full spectrum. Method can be validated as per ICH guidelines.¹³

The main objective of this work was to develop an additional method for the fast and accurate determination of Levosulpiride in commercial pharmaceutical formulations by using the Beer-Lambert law and reducing the sample pre-treatment and providing direct FT-IR measurement.

**Fig. 1: Chemical structure of levosulpiride**

**Experimental**
**Apparatus:** Data acquisition was performed using an FTIR spectrometer equipped with ATR, Zn-Se Crystal, OPUS Software (Bruker Co., German.). The commercial software used to generate analysis for the principal component analysis was Quant Builder.

**Reagents and Materials:** Levosulpiride pure sample was supplied by Sun Pharmaceuticals, Mumbai as gift sample and commercial tablets NEXIPRIDE
manufactured by Sun Pharmaceutical Industries Ltd, Mumbai, NEOPRIDE tablets manufactured by Intas Pharmaceuticals Ltd, Ahmedabad were procured from local market.

**Experimental Procedure:** Accurately weighed 50mg of Levosulpiride pure drug were kept in top of ATR crystal and spectra was recorded between 4000 and 650 cm\(^{-1}\), by averaging 24 scans for spectrum using OPUS software of Bruker-α /ZnSe FTIR spectrophotometer with Reflection Top-Plate.\(^{14}\) In order to compress the sample against the crystal, a pressure plate and clamp provided were used. One isolated peak 3159-3012 cm\(^{-1}\) was defined and peak integral area was calculated.

**Linearity and Range:** Different weights of Levosulpiride ranging from 50-100 mg were taken using electronic balance with 0.001 mg sensitivity and spectrum recorded and average of such three determinations were plotted in calibration curve (Fig. 2).

**LOD and LOQ:** Limit of detection (LOD) and Limit of quantitation (LOQ) for the assay was calculated using the following formula: LOD=3.3× (standard deviation of y-intercept of the regression line / slope of the calibration curve) LOQ = 10 × (standard deviation of y-intercept of the regression line / slope of the calibration curve)

**Assay of Levosulpiride Content in Tablet Dosage form:** Commercial samples of Tablet Nexipride and Neopride were taken and twenty such tablets from each company were powdered and suitable weights were taken and spectra recorded like that of standard. Peak area integration done for the same peak and quantitative analysis performed with Quant Builder option in the software.

**Accuracy:** Accuracy of the developed method was carried out by performing recovery study using standard addition method, in which standard drug was added at two different concentrations (50% and100%) to the pre-analyzed formulation

**Precision:** Precision study of the method was performed by intra-day and inter-day variation study. The intraday precision and inter-day precision was ascertained by determining AUC of 3 replicates of a fixed concentration of the drug (50mg) at three different time period of the same day and on three different days. The result of the precision studies was expressed in terms of % RSD (percentage of Relative Standard Deviation).

**Reproducibility:** Spectra were recorded for 50 mg of standard drug and it was repeated for six times.

**Results and Discussion**

**Linearity:** Levosulpiride was found to be linear within the concentration range 50-100 mg and exhibited correlation coefficient of 0.999 (Fig. 2).

**LOD and LOQ:** Stacked spectra of standard Levosulpiride (Fig. 3) and overlay spectra of standards (Fig. 4) were recorded to show the precision of the method. It is of interest to note that there are no significant differences in the fingerprinting region between the spectra obtained for standard and sample, a common peak between 3159 and 3012 cm\(^{-1}\) was selected for Integration (Fig. 5)

![Fit vs. True / Comp. 1 [mg]](image)

**Fig. 2:** Calibration graph of levosulpiride standards along with sample according to beer-lambert’s law
Fig. 3: Stacked spectra of levosulpiride standards

Fig. 4: Overlay spectra of levosulpiride standards
Fig. 5: Integration of standard levosulpiride

Statistical studies like Linearity, reproducibility, correlation coefficient, student’s t-test, accuracy and precision were recorded. (Table 1)

Table 1: Statistical analysis

| Parameter                  | Value                                      |
|----------------------------|--------------------------------------------|
| Linear regression          | $y = +189.04 + 152.39x$                    |
| Reproducibility            | $49.5 \pm 1.02 \text{ mg (}n=6\text{)}$    |
| Student’s t test           | $P = 0.05\%$ ($100.04 \pm 1.49\%, n = 3$) |
| Accuracy and precision in the range | $0.77-3.2\%$                          |
| $R^2$                      | $0.999$                                    |
| Linearity                  | $50-100 \text{ mg}$                       |

Table 2: Recovery studies

| Drug   | Amount of sample taken(mg) | Amount of standard added(mg) | Recovery at % | % Recovery |
|--------|---------------------------|-------------------------------|---------------|-----------|
| NEXIPRIDE | 50                     | 25                            | 50            | 99.8      |
|         | 50                       | 50                            | 100           | 98.6      |
| NEOPRIDE | 50                     | 25                            | 50            | 98.2      |
|         | 50                       | 50                            | 100           | 99.4      |

Table 3: Estimation of levosulpiride in tablet dosage form

| Drug        | Sample No | Concentration as per label claim(mg) | Amount found(mg) | Percentage label claim % | Mean %  |
|-------------|-----------|--------------------------------------|------------------|--------------------------|---------|
| NEXIRIDE    | 1         | 50                                    | 49.50            | 99.0                     | 100.7   |
|             | 2         | 50                                    | 51.02            | 102.4                    |         |
| NEOPRIDE    | 1         | 50                                    | 49.67            | 99.30                    | 100.45  |
|             | 2         | 50                                    | 50.80            | 101.60                   |         |

Conclusions

It is clear that FT-IR spectrometry is capable of direct determination of Levosulpiride in formulations. With the commercial software involving chemometric approaches, Beer-Lambert law, the method proposed is simple, precise and not time-consuming compared to the chromatographic methods that exist in literature. Quantification could be done in about 5-10 minutes, including sample preparation and spectral acquisition.
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