Metal Transformation by a Novel Pelosinus Isolate From a Subsurface Environment

Allison E. Ray1,2, Stephanie A. Connon1,3, Andrew L. Neal4†, Yoshiko Fujita2, David E. Cummings5, Jani C. Ingram2† and Timothy S. Magnuson1*

1 Department of Biological Sciences, Idaho State University, Pocatello, ID, United States, 2 Idaho National Laboratory, Idaho Falls, ID, United States, 3 California Institute of Technology, Pasadena, CA, United States, 4 Savannah River Ecology Laboratory, University of Georgia, Aiken, SC, United States, 5 Department of Biology, Point Loma Nazarene University, San Diego, CA, United States

The capability of microorganisms to alter metal speciation offers potential for the development of new strategies for immobilization of toxic metals in the environment. A metal-reducing microbe, “Pelosinus lilae” strain UFO1, was isolated under strictly anaerobic conditions from an Fe(III)-reducing enrichment established with uncontaminated soil from the Department of Energy Oak Ridge Field Research Center, Tennessee. “P. lilae” UFO1 is a rod-shaped, spore-forming, and Gram-variable anaerobe with a fermentative metabolism. It is capable of reducing the humic acid analog anthraquinone-2,6-disulfonate (AQDS) using a variety of fermentable substrates and H2. Reduction of Fe(III)-nitritriacetic acid occurred in the presence of lactate as carbon and electron donor. Ferrihydrite was not reduced in the absence of AQDS. Nearly complete reduction of 1, 3, and 5 ppm Cr(VI) occurred within 24 h in suspensions containing 10^8 cells mL^-1 when provided with 10 mM lactate; when 1 mM AQDS was added, 3 and 5 ppm Cr(VI) were reduced to 0.1 ppm within 2 h. Strain UFO1 is a novel species within the bacterial genus Pelosinus, having 98.16% 16S rRNA gene sequence similarity with the most closely related described species, Pelosinus fermentans R7T. The G+C content of the genomic DNA was 38 mol%, and DNA-DNA hybridization of “P. lilae” UFO1 against P. fermentans R7T indicated an average 16.8% DNA-DNA similarity. The unique phylogenetic, physiologic, and metal-transforming characteristics of “P. lilae” UFO1 reveal it is a novel isolate of the described genus Pelosinus.

Keywords: toxic metal reduction, bioremediation, subsurface environment, fermentative bacterium, Pelosinus

INTRODUCTION

The pollution of groundwater by metal and radionuclide contaminlants continues to pose a clear threat to public health (EPA, 2003), and remediation of subsurface environments is a growing environmental and economic challenge (White et al., 1997; NABIR, 2003). In the United States, the Department of Energy (DOE) has generated 6.4 trillion L of contaminated groundwater in 5700 distinct plumes and 40 million m^2 of contaminated soil and subsurface environs as a result of Cold War era nuclear weapons production (USDOE, 1997). After
petroleum fuel and chlorinated hydrocarbons, metals and radionuclides (lead, chromium, arsenic, uranium, strontium-90, and tritium) represent the most common contaminants found in soils and groundwater at DOE facilities (Riley et al., 1992). Moreover, metallic and radioactive contaminants cannot be biodegraded and often prove to be the most challenging of hazardous wastes at DOE sites (NABIR, 2003).

In contrast to organic contaminants that may be completely oxidized to carbon dioxide and water, remediation of metals is more problematic, owing to the fact that the metals cannot be destroyed, and changes in their oxidation or reduction potential often dictate their ultimate environmental fate (Lovley and Lloyd, 2000). Generally, the goal of metal remediation is to limit contaminant mobility. The potential for immobilization of a given metal in the environment is principally dependent upon its chemical speciation. Thus, the aim is to either induce or maintain chemical conditions that result in metal species with reduced mobility. Depending on the ambient redox conditions, many toxic metals and radionuclides can be highly soluble, and thus mobile, in groundwater (Lovley, 2001). Microorganisms are capable of altering chemical speciation via redox reactions, thereby influencing solubility, transport properties, and bioavailability of metallic contaminants in subsurface environments. For example, bioreduction of highly soluble Cr(VI) and U(VI) can result in conversion to insoluble species [e.g., Cr(III) and U(IV)] that precipitate from solution (Tabak et al., 2005).

Microbial Fe(III) reduction is an important environmental process, and many Fe(III)-reducing bacteria have been found to also reduce high-valence metal contaminants (Lloyd, 2003). In addition, Fe(III)-reducing bacteria have been shown to utilize humic acids and synthetic electron-shuttling moieties, such as anthraquinone-2,6-disulphonate (AQDS), as electron acceptors (Lovley et al., 1996; Scott et al., 1998), which in turn can mediate the indirect reduction of Fe(III) and other metals (Lovley et al., 1998; Fredrickson et al., 2000; Nevin and Lovley, 2000). The physiologic diversity of electron transport to ferruginous mineral substrates is distributed into distinct groups of Fe(III)-reducing microorganisms. One group consists of the respiratory Fe(III)-reducers, in which reduction occurs via an electron transfer event coupled to energy generation. Representatives of this group include well-characterized organisms such as Geobacter spp., and Shewanella spp. (Lovley et al., 1987, 1989, 1993; Caccavo et al., 1992, 1994, 1996). The second group includes fermentative microorganisms that use humic acids and Fe(III)-bearing minerals as an electron sink for excess reducing power formed during fermentative metabolism (Benz et al., 1998; Borch et al., 2005; Shelobolina et al., 2007). Although these organisms have been detected in a variety of habitats where Fe(III) reduction is an important ecophysiological process (Kappler et al., 2004), this group is less well understood physiologically and phylogenetically. Recent studies on isolated members of the genus *Pelosinus* reveals novel mechanisms for metal sequestration and transformation (Beller et al., 2013; Thorgersen et al., 2017), and exhibit unique phylogenomic traits such as multiple, distinct copies of 16S rRNA genes (Ray et al., 2010), potentially confounding phylogenetic analysis of this group common in subsurface environments. Thus, examining new representatives of this group contributes to a broader understanding of the organisms involved in the biotransformation of metals in subsurface environments.

In situ biological treatment can be less expensive and less disruptive than traditional ex situ technologies for remediation of metal-contaminated sites, as it relies on indigenous microorganisms to achieve clean-up of hazardous wastes (NABIR, 2003). Examination of the biological potential for metal reduction among native microorganisms is important for implementation of successful remediation strategies. However, little is known about the potential for fermentative, Fe(III)-reducing subsurface microorganisms to play a role in metal bioremediation. Our goal was to isolate and characterize one such organism from a field study area adjacent to a metals-contaminated environment and evaluate its potential for metal bioremediation. Here, we describe "*Pelosinus lilae*" UFO1, isolated from the background or “pristine” area at the Oak Ridge Field Research Center (ORFRC), Tennessee. The unique phylogenetic and metal-transforming characteristics of “*P. lilae*” UFO1 reveal it is a novel species of the described genus designated *Pelosinus*.

### MATERIALS AND METHODS

#### Enrichment and Isolation of Strain UFO1

Strict anaerobic techniques were used throughout this study (Miller and Wolin, 1974; Balch and Wolfe, 1976). Sediment cores from the Field Research Center in Oak Ridge, TN, United States were taken from background well 330, section 02-22 (FWB 330-02-22), at a depth of 0.61–1.12 m below the surface. The groundwater pH was 6.13. Samples were shipped and stored in Mason jars under N2 and processed in an anaerobic glove bag under N2-CO2-H2 (75:20:5).

Enrichment cultures were initiated in 27 mL anaerobic pressure tubes (Bellco Glass) containing 9 mL anaerobic freshwater medium (ATCC medium 2129) and ~1 g FRC sediment. A suspension of 2-line ferrihydrite (~1 M) was synthesized by dissolving 40 g Fe(NO3)3·9H2O in 0.5 L water and adjusting to pH 7.0 with ~345 mL 1 M KOH, and the resulting precipitate was washed thoroughly in DI water (Schwertmann and Cornell, 2000). The ferrihydrite was used as the terminal electron acceptor for enrichment cultures. Ferrihydrite (~40 mM) was added to the tubes along with 10 mM acetate, and the tubes were incubated at room temperature. After 9 months, a 30 mM bicarbonate-buffered freshwater medium with 10 mM lactate, 5 mM AQDS as described by Finneran et al. (2002), and ~20 mM ferrihydrite was used to prepare a dilution series from the iron-reducing culture. Aliquots were removed from the highest dilutions appearing positive for AQDS- and Fe(III)-reduction and streaked on plates of R2A agar (BD DifcoTM, Franklin Lakes, NJ, United States) supplemented with 20 mM fumarate in an anaerobic chamber. Distinct colonies were picked and re-streaked at least three times prior to transfer to liquid medium.
**Routine Cultivation**

After isolation, "P. lilae" UFO1 was routinely cultured in anoxic R2A broth and incubated at 30°C. R2A broth was prepared from a dry mix (BD Diagnostic Systems, Franklin Lakes, NJ, United States) or as follows (per liter): 0.5 g yeast extract; 0.5 g proteose peptone; 0.5 g casamino acids; 0.5 g glucose; 0.5 g soluble starch; 0.3 g sodium pyruvate; 0.3 g K₂HPO₄; 0.05 g MgSO₄ (Reasoner and Geldreich, 1985), supplemented with 20 mM fumarate, and adjusted to pH 7. The R2A broth was boiled and cooled under a headspace of N₂, dispensed into anaerobic pressure tubes or serum vials with a headspace of N₂, sealed with thick butyl-rubber stoppers, and autoclaved. The effects of temperature, pH, and O₂ on growth rates were evaluated using R2A broth (adjusted to acidic and alkaline pH where necessary).

Anaerobic, bicarbonate-buffered, freshwater (FW) medium was prepared, as described previously by Finneran et al. (2002) with or without 5 mM AQDS, and dispensed into anaerobic pressure tubes or serum vials under N₂-CO₂ (80:20). Tubes or vials were sealed with thick butyl-rubber stoppers and sterilized by autoclaving. FW medium was used for experiments examining electron donor and acceptor utilization.

**Electron Donor and Acceptor Utilization Studies**

Cells were harvested by centrifugation from cultures grown in R2A broth, washed twice in anoxic FW medium, and suspended in sterile, pH 7, anoxic FW medium. For screening of potential electron donors and acceptors, sterile anoxic stock solutions were prepared for a variety of electron donors and acceptors and added to the FW medium. Utilization was evaluated in triplicate incubations.

The ability of "P. lilae" UFO1 to reduce the following electron acceptors was examined: Fe(III)-NTA, 2 line-ferrihydrite, AQDS, Cr(VI), As(V), NO₃⁻, NO₂⁻, SO₄²⁻, and SeO₄²⁻. The ability of washed cell suspensions to reduce Cr(VI) was examined under non-growth conditions, defined here by the omission of phosphate from the FW medium. Washed cells were added to the medium to give a concentration of 10⁶ cells mL⁻¹.

Carbon substrate utilization by strain UFO1 was examined under anaerobic conditions using the BIOLOG® AN MicroPlate™ (Hayward, CA, United States) (Bochner, 1989). Anaerobic MicroPlates test the ability of a microorganism to utilize or oxidize an array of carbon sources under anaerobic conditions using an artificial tetrazolium colorimetric electron acceptor. A culture of strain UFO1 was grown for 24 h on R2A broth, pelleted and washed in anoxic FW medium, and suspended in the inoculating medium according to the manufacturer’s instructions. Microplates were inoculated in triplicate and incubated at 35°C for 24 h; wells exhibiting a color change due to the reduction of the tetrazolium dye as compared to the negative control (no carbon source) were considered positive for substrate utilization.

Potentially fermentable substrates were examined in FW medium (containing 0.5 mM cysteine as a reducing agent) in the absence of an electron acceptor. The following substrates (10 mM each) were tested: fructose, fumarate, glucose, glycerol, lactate, malate, mannitol, pyruvate, succinate, and 0.3% wt/vol. yeast extract. Fermentation of substrates was defined as the ability to grow after three successive 10% transfers, and growth was monitored by measuring optical density at 600 nm.

**Analytical Techniques**

AQDS and anthrahydroquinone-2,6-disulfonate (AH₂DS) were measured spectrophotometrically at 405 and 325 nm under anoxic conditions. The molar absorptivity coefficients calculated for AQDS were ε₄₀₅ = 0.13 and ε₃₂₅ = 6.1 mM⁻¹ cm⁻¹; for AH₂DS, the coefficients used were ε₄₀₅ = 10.3 and ε₃₂₅ = 0.8 mM⁻¹ cm⁻¹. Aqueous Fe(II) in sample filtrate (filtered through 0.2-µm filter) that was diluted 1:10 in 0.5 N HCl was quantified spectrophotometrically at 560 nm with ferrozine (Lovley and Phillips, 1987). Diphenylcarbazide reagent (Hach, Loveland, CO, United States) was used to quantify Cr(VI) at 540 nm. Ion chromatography (IC25 Ion Chromatograph Ion Pac-ATC-HC Trap Column) was used to measure NO₃⁻, NO₂⁻, SO₄²⁻, and SeO₄²⁻. Reduction of As(V) was evaluated by measuring soluble As(V) at 865 nm in samples preserved in KIO₃ and HCl as described previously (Cummings et al., 1999).

X-ray photoelectron spectroscopy (XPS) was employed to determine the valence state of Fe in cell suspensions incubated with 5 mM AQDS and 2 line-ferrihydrite or ferrihydrite alone. The extent of Fe(III)-reduction at both uncolonized and colonized Fe(III)-oxide surfaces was examined. Ferrihydrite samples were dried onto a Si wafer before mounting for XPS analysis. Samples were mounted for XPS in an anaerobic glove box and transported to the spectrometer in a plate chamber sealed under anaerobic atmosphere. Brief (<5 s) exposure to air occurred during introduction of the samples into a N₂-flushed antechamber. This antechamber was then evacuated and the sample placed into the spectrometer itself. Spectra were collected on a Perkin–Elmer Physical Electronics Division Model 5600ci spectrometer (Perkin–Elmer Inc., Eden Prairie, MN, United States). The spectrometer was calibrated employing the Au 4f₇/₂, Cu 2p₁/₂, and Ag 3d₃/₂ photopeaks with binding energies of 83.99, 932.66, and 368.27 eV, respectively. A consistent 400-µm spot size was analyzed on all surfaces using a monochromatized Al Kα (hv = 1486.6 eV) X-ray source at 300 W and a pass energy of 46.95 eV for survey scans, and 11.75 eV for high-resolution scans. The system was operated at a base pressure of 10⁻⁸ to 10⁻⁹ torr. An emission angle (2θ) of 45° was used throughout. Following baseline subtraction (Shirley, 1972), curve-fitting was employed using an 80% Gaussian: 20% Lorentzian line shape to estimate the contributions of ferrous and ferric ions to the total Fe photopeak (described below). A common problem associated with the analysis of insulating materials such as Fe(III)-oxides is the accumulation of surface charge during spectral collection leading to photopeak shifts; this was overcome by the use of a 5-eV flood gun and by referencing of the principal C 1s photopeak [namely due to carbon of the type (–CH₂–CH₂–)ₙ] to a binding energy (E₀) of 284.8 eV (Swift, 1982).
Theoretical core p level models describing multiplet splitting associated with ferrous and ferric ions have been demonstrated by Gupta and Sen (Gupta and Sen, 1975) and empirical proofs substantiated by Pratt et al. (1994). Accordingly, ferric ion contributions were fit to high-resolution Fe 2p \(_{1/2}\) core regions with five peaks that decrease in intensity at increased \(E_b\), with all peaks having the same line-shape. Ferrous ion contributions are represented by a major peak accompanied by a pair of multiplet peaks occurring 0.9 eV to either side of the major peak and a shake-up satellite at elevated \(E_b\) [\(\Delta E_b\approx 6\) eV, (McIntyre and Zetaurk, 1977)]. The relative intensities of the Fe(II) peaks are consistent with the aforementioned models and have been applied elsewhere in the identification of reduced iron in the presence of microorganisms (Herbert et al., 1998; Neal et al., 2001, 2004; Magnuson et al., 2004).

**DNA Base Composition and Cell Wall Analysis**

Cells were grown overnight on R2A broth supplemented with 20 mM fumarate at 30°C and harvested by centrifugation. Cell pellets were suspended in a mixture of isopropanol and water (1:1). The G+C content of genomic DNA was determined by high-performance liquid chromatography using the method of Mesbah et al. (1989), and peptidoglycan was analyzed by the method of Schleifer and Kandler (1972). Both analyses were performed at Deutsche Sammlung von Mikroorganismen und Zellkulturen (DSMZ).

**DNA–DNA Hybridization**

Cells of "P. lilae" UFO1 were grown in anoxic R2A broth at 30°C overnight and harvested by centrifugation. The pellet was suspended in a mixture of isopropanol and water (1:1) and submitted to DSMZ for DNA–DNA hybridization with *Pelosinus fermentans* R7\(^T\) (DSM 17108). DNA was isolated from the cell pellet using a French pressure cell and purified by chromatography on hydroxyapatite as described by Cashion et al. (1977). DNA–DNA hybridization was carried out as described previously (DeLey et al., 1970) with the modifications described by Huss et al. (1983) using a model Cary Bio UV/VIS-spectrophotometer equipped with a Peltier-thermostatted 6 × 6 multicell changer and a temperature controller with in situ temperature probe (Varian). Percent DNA–DNA similarity was measured in 2X saline sodium citrate (SSC) solution at 65°C.

**Fatty Acid Methyl Ester Analysis**

An overnight culture of "P. lilae" UFO1 was grown on R2A broth containing 20 mM fumarate at 30°C. The cells were harvested by centrifugation, frozen (−20°C), and analyzed by Microbial ID (Newark, DE, United States) for fatty acid methyl ester content.

**Electron Microscopy**

Cells were grown overnight at 30°C in R2A broth containing 20 mM fumarate for morphological characterization using a Philips XL 30 Environmental Scanning Electron Microscope (ESEM), operating at 10 kV with a typical target current of 1.75 μA. Prior to imaging, the cells were washed twice and re-suspended in phosphate buffered saline (pH 7). Droplets of the washed cell suspension were placed in a conical ESEM mount to maximize solution volume. The chamber was maintained at ~45% relative humidity (20°C, 2.4 torr of H₂O) using a combination of Peltier cooling and differential pumping. ESEM imaging looked first at the edges of the mount then progressed inward to find regions of optimal cell density, minimal desiccation, and minimal precipitation of salts.

**16S rRNA Gene Amplification**

Genomic DNA was extracted from "P. lilae" UFO1 using the Mo Bio Ultra Clean Microbial DNA kit (Mo Bio Laboratories, Inc., Carlsbad, CA, United States). PCR amplification of the 16S rRNA gene was performed with primers 8F (5′-AGAGTTTGATCCTGGCTCAG-3′) (Eden et al., 1991), 907R (5′-CCGTCATTCTMTTTRAGTTT-3′) (Lane et al., 1985), 704F (5′-GTAGCCGTTGAAATGCCTAG-3′) (Lane et al., 1985), and 1492R (5′-GGTTACCTTGTAGAACGCTT-3′) (Lane, 1991). Each 50 μL PCR reaction mixture contained 1 U mL⁻¹ of Deep Vent polymerase and 1X ThermoPol reaction buffer (New England Biolabs, Ipswich, MA, United States), 250 μM dNTPs, 800 nM of each primer, 2 mM MgSO₄, and 1.5 μL of genomic DNA template. The PCR amplification conditions were as follows: an initial 95°C denaturation for 5 min, followed by 30 cycles of 95°C denaturation for 1 min, 56°C primer annealing for 1 min, and 72°C extension for 2.5 min, and then a final extension at 72°C for 4 min. PCR products were purified using the Qiagen QIAquick PCR Purification Kit (Valencia, CA, United States) per manufacturer’s instructions. The 16S rRNA gene PCR products were sequenced at the Idaho State University Molecular Research Core Facility (MRCF) on an ABI 3100 automated capillary sequencer (Applied Biosystems, Foster City, CA, United States) using 8F, 519R (5′-ATTACCGCGGCTGCTGG-3′) (Lane et al., 1985), 907R, 704F, 1100F (5′-CAACGAGCAGCAACCC-3′) (Lane et al., 1985), and 1492R primers in order to guarantee overlap of sequences. PCR products amplified with 8F and 1492R primers were also cloned in order to resolve the 3’ and 5’ ends of the 16S rRNA gene. Immediately following the PCR reaction, 3′-A overhangs were added to blunt-end 16S rRNA gene amplicons generated with 8F/1492R and Deep Vent polymerase using Taq polymerase (Invitrogen, Carlsbad, CA, United States) in an additional 10-min extension performed at 72°C. For addition of 3′-A overhangs, the 25 μL reaction contained the following: 16.2 μL of genomic DNA, 1X buffer, 2 mM MgCl₂, 400 μM dATP, and 1 U Taq polymerase. The resulting 16S rRNA gene product was cloned into the pCR4 *-TOPO* vector using the TOPO TA Cloning® Kit for Sequencing (Invitrogen, Carlsbad, CA, United States) according to the manufacturer’s instructions. Plasmids were purified from clones using the Qiagen Qiaquick Spin Miniprep kit (Valencia, CA, United States) and sequenced directly using primers M13F (5′-GTAAAAACGAGCCAG-3′), M13R (5′-CAGGAAACGCTTACG-3′), T3 (5′-ATTACCC TCACTAAAGGGG-3′), and T7 (5′-TATACGACTCACTATA GGG-3′) at the ISU MRCF. The GenBank accession number for
the 16S rRNA sequence of “*P. lilae*” UFO1 is DQ295866, and the draft genome accession number is CP000852.

**Phylogenetic Analysis**

Species representing the Negativicutes Class were determined based on the List of Prokaryotic Names with Standing in Nomenclature (LPSN)\(^1\) (Parte, 2014). Phylogenetic analysis was conducted using both ARB, version 6.0.6 (Strunk et al., 1996) software, and the SILVA sequence database SSURef_NR99_128_SILVA_07_09_16_opt\(^2\) (Quast et al., 2013). Sequences not available in this SILVA database were downloaded from GenBank\(^3\). SILVA sequence alignments were used with a few refinements made manually. Sequences were filtered using SILVAs positional variability by parsimony filter for bacteria (pos_var_ssuref:bacteria). Phylogenetic tree was inferred within the ARB software by the Maximum Likelihood method PHYML, version 20130708 (Guindon and Gascuel, 2003), using nucleotide substitution model HKY85 (Hasegawa et al., 1985) and branch supports determined by Bayesian estimation.

**RESULTS AND DISCUSSION**

### Strain Isolation and Morphology

*“Pelosinus lilae”* UFO1 was isolated under strictly anaerobic conditions from dilution-to-extinction cultivation experiments, and was capable of AQDS reduction, as evident by a color change in growth medium from transparent, pale yellow to bright orange. An aliquot of enrichment culture was streaked on R2A plates containing 20 mM fumarate, and after 48–72 h of anaerobic incubation, small, round, white colonies were apparent. After re-streaking at least three times, a colony was transferred to FW medium containing lactate and AQDS, and the culture reduced AQDS. “*P. lilae*” UFO1 is a strict anaerobe with a fermentative metabolism, and microscopic observations revealed that it is rod-shaped, 0.2–0.7 × 1.5–4.7 μm in size (Figure 1-left panel), motile, and stains Gram-variable. Structures that appeared to be spores were also visible (Figure 1-right panel). Extracellular polymeric substance (EPS) is associated with individual cells, as judged by electron-dense materials surrounding the cell periphery. The production of spores was supported by the observation that a 10% inoculum from thermally-treated (85°C for 32 min) cell suspensions of “*P. lilae*” UFO1 could be re-grown in R2A broth and also exhibited the ability to reduce AQDS in the presence of H₂ upon transfer to fresh media.

### Temperature and pH Tolerance

A growth curve for “*P. lilae*” UFO1 at 30°C is shown in Figure 2. Optimal temperature and pH for growth of “*P. lilae*” UFO1 were determined using R2A broth supplemented with 20 mM fumarate. The range of growth temperatures was 22–37°C, with an optimum of 37°C; average generation times at 22, 30, and 37°C were 4.6, 2.9, and 1.9 h, respectively. The range of pH for growth on R2A broth with 20 mM fumarate was 5.5–8, with optimum growth at pH 7. No growth was observed at temperatures of 13°C or 42°C, or at pH ≤ 5.4 or pH ≥ 8.5.

### Electron Donor Utilization

Carbon sources utilized in Biolog AN microplates were D-fructose, L-fucose, D-galactose, D-galacturonic acid, D-glucose-6-phosphate, glycerol, 3-methyl-D-glucose, palatinose, L-rhamnose, α-ketobutyric acid, α-ketovaleric acid, pyruvic acid, and succinic acid. “*P. lilae*” UFO1 fermented the following substrates (as evidenced by growth in the absence of an externally supplied electron acceptor): fructose, fumarate, glucose, glycerol, lactate, mannitol, pyruvate, and yeast extract. Malate and succinate were tested but not fermented. “*P. lilae*” UFO1 did not grow chemosmotrophically on H₂-CO₂ (see Table 1).

### Metal Transformation

Although “*P. lilae*” UFO1 was isolated from an Fe(III)-reducing enrichment by cultivation with acetate and AQDS, this “*P. lilae*” was not capable of respiratory growth on AQDS when 10 mM acetate was provided as the electron donor. AQDS was reduced to AH₂DS only in the presence of fermentable substrates or H₂ (Table 1). In the presence of 10 mM lactate, nitrate was reduced to nitrite following 1 week of incubation. As(V), NO₃⁻, SeO₄²⁻, and SO₄²⁻ were not reduced in the presence of 10 mM lactate.

The incomplete reduction of 10 mM Fe(III)-NTA by “*P. lilae*” UFO1 is shown in Figure 3. In the presence of 10 mM lactate, nearly 3 mM Fe(II) was produced after 1 week of incubation. Production of Fe(II) was not evident in abiotic controls. Biotic reduction of Fe(III)-NTA in the absence of lactate did not occur.

High-resolution core Fe 2p\(_3/2\) spectra collected on a cell-free control containing ferrihydrite incubated with culture medium and H₂ suggested that abiotic Fe(III)-reduction did not take place under these conditions (Table 2). The collected peak envelope is described well by the Fe(III) multiplet splitting model and is in agreement with published iron oxide spectra (data not shown). However, cell-free controls incubated with H₂ + AQDS, lactate, or lactate + AQDS revealed the presence of 14.8, 26.2, and 26.0 atom% Fe(II), respectively, indicating some abiotic reduction of ferrihydrite occurred. Incubation of ferrihydrite with strain UFO1 and H₂, both with and without AQDS, revealed a significant increase in the Fe(II) contribution compared to the abiotic controls. Addition of a principal Fe(II) peak at 708.4 eV for the ferrihydrite spectra (together with associated multiplet and satellite peaks) was required to complete the model (not shown). In contrast, lactate did not appear to significantly enhance reduction of ferrihydrite by “*P. lilae*” UFO1 regardless of the presence of AQDS.

Soluble Cr(VI) was removed from solution in cell suspensions of “*P. lilae*” UFO1 under non-growth conditions. Figure 4A shows the nearly complete removal of 1 and 3 ppm Cr(VI) by strain UFO1 within 24 h when lactate (10 mM) was provided as an electron donor, whereas 5 ppm Cr(VI) decreased to an average 0.3 ppm in 26 h. Over the course of the experiment, controls containing heat-treated cells (90°C, 32 min) did not exhibit significant Cr(VI) removal suggesting that sorption to cell surfaces did not contribute to the removal of Cr(VI) seen in

---

\(^1\)www.bacterio.net
\(^2\)www.arb-silva.de
\(^3\)www.ncbi.nlm.nih.gov
other treatments. In Figure 4B, it is evident that Cr(VI) removal was dramatically enhanced in the presence of 1 mM AQDS. For treatments containing 3 and 5 ppm Cr(VI) with lactate (10 mM) plus 1 mM AQDS, Cr(VI) concentrations dropped to less than 0.1 ppm after just 2 h of incubation with cell suspensions of strain UFO1. There was no significant change in Cr(VI) concentrations in cell-free and heat-treated controls. In contrast to the Fe(III)-NTA results, where lactate appeared to be required for Fe(III) reduction, Cr(VI) was removed in the absence of lactate from 3 to 0.5 ppm in 24 h, which may suggest the use of endogenous carbon reserves by “P. lilae” UFO1, if removal is assumed to be the result of reduction. Others have similarly observed removal of soluble Cr(VI) by other bacterial species in the absence of exogenous electron donors (Ishibashi et al., 1990; Shen and Wang, 1993; Sani et al., 2002), suggesting that Cr(VI) removal occurs via the action of an enzyme utilizing endogenous electron donors (i.e., cellular NADH, FADH$_2$) with a non-specific NADH oxidoreductase activity that can reduce Cr(VI) to Cr(III). In Pelosinus sp. HCF1 (Beller et al., 2013), Cr reduction was thought to be linked to a flavoprotein related to the ChrR family of chromate reductases, with potential involvement of hydrogenases. Genome analysis suggests a similar potential in “P. lilae” UFO1 (see below).

“P. lilae” UFO1 was previously evaluated for its U(VI) transformation abilities (Ray et al., 2011), and a potential mechanism for U(VI) sequestration in this isolate was discovered (Thorgersen et al., 2017). Interactions of U(VI) with extracellular materials led to the discovery of S-layer mediated binding of U(VI), as well as some reduction activity [appearance of U(IV)]. While the precise mechanisms of sequestration and reduction of Cr, U, and Fe in strain UFO1 remain poorly characterized, we propose that a combined sequestration/transformation mechanism is at play for dealing with Cr, U, Fe, and perhaps a variety of other metals not yet evaluated. Previous studies examining S-layer metal binding in Bacillus support this idea (Velásquez and Dussan, 2009), but the true functionality of

| Electron donor | AH$_2$DS (mM) |
|----------------|---------------|
| Acetate (10 mM) | 0.03 ± 0.00 |
| Benzoate (0.5 mM) | 0.04 ± 0.01 |
| Ethanol (10 mM) | 0.04 ± 0.01 |
| Formate (10 mM) | 0.05 ± 0.01 |
| Glucose (5 mM) | 1.05 ± 0.01 |
| Glycerol (10mM) | 1.01 ± 0.48 |
| H$_2$ (5 ml in headspace) | 2.52 ± 0.17 |
| Phenol (0.5 mM) | 0.04 ± 0.01 |
| Propionate (10 mM) | 0.05 ± 0.01 |
| Pyruvate (10 mM) | 1.47 ± 0.03 |
| Succinate (10 mM) | 0.22 ± 0.10 |
| Yeast Extract (0.25%) | 2.20 ± 0.10 |

Average production of AH$_2$DS in triplicate cultures (±1 standard deviation) after 7 days of incubation at 30°C with “P. lilae” UFO1.
“P. lilae” UFO1 S-layer proteins in toxic metal binding and sequestration remains to be investigated.

Biochemical Characteristics

The majority of fatty acids identified in strain UFO1 were straight unsaturated chains in cis conformation (Table 3). The predominant fatty acids identified were C15:1 ω8c, C17:1 ω8c, and C15:0. “P. lilae” UFO1 had straight saturated (35.52%) and unsaturated (62.38%) chains, 6.59% C11 and C13 fatty acids, 74.87% C15 and C17. The fatty acid profile of strain UFO1 is consistent with profiles characteristic of members of the Class Negativicutes described previously (Strompl et al., 1999) with the exception of 3-hydroxy fatty acids, which were only present at 1.82%. Dimethyl acetal (C14:0 DMA and C14:1 ω7c DMA) accounted for 9.52% of the total fatty acids identified and are characteristic of anaerobic bacteria. Significant amounts of dimethyl acetal among members of the Class Negativicutes have been reported previously (Moore et al., 1994).

The cell wall of “P. lilae” UFO1 contained mesodiaminopimelic acid (m-Dpm) as the diagnostic diamino acid in the total hydrolysate of the peptidoglycan. Alanine and glutamic acid were also present in the peptidoglycan. Partial hydrolysis of the peptidoglycan revealed the presence of the peptides L-ala—D-glu and Dpm—D-ala. From these data, it was concluded that “P. lilae” UFO1 shows the directly cross-linked peptidoglycan type, A1γ m-Dpm-direct (Schleifer and Kandler, 1972).

The G+C content of the genomic DNA of “P. lilae” UFO1 was 38.0 mol% (Table 4). The DNA G+C content of P. fermentans R7T, is 41.0 mol% (Shelobolina et al., 2007). Additionally, duplicate DNA-DNA hybridizations conducted with strain UFO1 against P. fermentans R7T showed 9.8 and 23.7% DNA-DNA similarity, indicating that strain UFO1 does not belong to the species P. fermentans as defined by the threshold value of 70% DNA-DNA relatedness (Wayne et al., 1987).

16S rRNA Gene Sequence and Phylogenetic Analysis

“Pelosinus lilae” UFO1 is firmly included within the Pelosinus clade in the Sporomusaceae family with branch support of 100 (Figure 5). Results indicate that “P. lilae” UFO1 is phylogenetically distinct from the most closely related organisms P. fermentans R7T (Shelobolina et al., 2007), Pelosinus propionicus
TABLE 3 | Equivalent chain length (ECL) and fatty acid composition (%) of “P. lilae” UFO1 and comparator strains.

| ECL | Fatty Acid | 1 % | 2 % | 3 % |
|-----|------------|-----|-----|-----|
| 9.00 | C18:0 | 0.79 | 1.7 | 1.4 |
| 10.00 | C16:0 | 0.28 | 1.8 | 2.8 |
| 10.61 | i-C11:0 | 1.25 | 4.0 | 3.6 |
| 11.00 | C12:0 | 2.82 | 3.6 | 4.9 |
| 13.00 | C13:0 | 1.67 | 0.5 | 1.0 |
| 13.46 | C12:0 3OH | 0.97 | ND | ND |
| 14.00 | C14:0 | 0.72 | 0.7 | 1.9 |
| 14.11 | i-C13:0 3OH | 0.85 | 4.6 | 4.2 |
| 14.28 | C14:1 i7c DMA | 1.27 | 1.3 | 0.9 |
| 14.47 | C14:0 DMA | 8.25 | 10.0 | 12.4 |
| 14.79 | C15:1 i8c | 28.45 | ND | ND |
| 14.85 | C15:1 i6c | 0.99 | 1.4 | 0.8 |
| 15.00 | C15:0 | 13.45 | 5.9 | 5.5 |
| 15.77 | C16:1 i8c | 2.04 | 2.4 | 4.7 |
| 15.81 | C16:1 i7c | 1.09 | 2.3 | 2.2 |
| 16.00 | C16:0 | 1.92 | 1.3 | 1.9 |
| 16.79 | C17:1 i8c | 26.52 | ND | ND |
| 16.86 | C17:1 i6c | 1.53 | 1.2 | 0.7 |
| 17.00 | C17:0 | 3.93 | 1.3 | 0.5 |
| 17.77 | C18:1 i8c | 0.49 | 0.3 | 0.9 |
| 18.00 | C18:0 | 0.72 | ND | ND |

1 Pelosinus strain UFO1, 2 Pelosinus defluvii SHI-1T, 3 Pelosinus fermentans DSM 17108T.

strains TmPN3T and TmPM3 (originally published as Sporotalea propionica) (Boga et al., 2007), and Pelosinus defluvii (Moe et al., 2012). The distances between 16S sequences from “P. lilae” UFO1 and the closest type strains, P. fermentans strain R7T and P. propionicus TmPN3T, were 1.84 and 2.40% respectively, which are greater than the distance between the described type strains, 0.9%. A BLAST search of the 16S rRNA gene sequence of strain UFO1 revealed 99% similarity (over 1041 nucleotide bases) with a clone detected in a pH 5, Fe(III)-reducing enrichment established with background sediments from the FRC, pHSlac302-37 (AY527741) (Petrie et al., 2003). Additionally, two clones detected by Petrie and co-workers (Petrie et al., 2003) that were from Fe(III)-reducing enrichments established with contaminated FRC sediments also shared a high degree of 16S rRNA gene sequence similarity with “P. lilae” UFO1: 97% for Gly030-8A (AY524569) and 97% for Gly030-5C (AY524568). A recent paper by Newsome et al. (2015) report the presence of Pelosinus in enrichments of sediments from a United Kingdom nuclear site. Stimulation with glycerol phosphate resulted in substantial increases in bacteria closely related to Pelosinus, which comprised 33% of bacteria identified at the genus level. This work implicates Pelosinus species as having a key role in the removal of soluble U(VI) via precipitation to a reduced, crystalline U(IV) phosphate mineral, considered to be more recalcitrant to oxidative remobilization, in contaminated sediments.

During the analysis of the 16S rRNA gene sequence of “P. lilae” UFO1, significant inter-operon heterogeneity among 16S rRNA gene clones from “P. lilae” UFO1 was detected. Sequencing of cloned 16S rRNA gene PCR products revealed two distinct 16S rRNA gene sequences were present in strain UFO1. These differences were not due to the presence of a contaminant, as the culture was extensively purified prior to analysis (Ray et al., 2010). In one group of clones a 100-bp insertion was present near the 5’ end of the sequence. This type of 16S rRNA gene sequence heterogeneity was previously reported for the, proposed but not validly published, member of the Class Negativicutes “Anaerospora hongkongensis” (Woo et al., 2005; Beller et al., 2013). 16S rRNA length heterogeneity has also been reported in a few other unrelated species, notably Paenibacillus polyxymyx (Nubel et al., 1996), Desulfitomaculum kuznetsovi (Tourova et al., 2001), Aeromonas strains (Morandi et al., 2005), Bacillus claussii (Kageyama et al., 2007), and the archaeon Haloarcula marismortui (Mylvaganan and Dennis, 1992; Amann et al., 2000). Further work has demonstrated that the insert-bearing 16S rRNA gene sequence was not functional in the ribosomes of strain UFO1 (Ray et al., 2010).

Genome Annotation and Implications for Metal Transformation

The draft genome sequence for strain UFO1 (Brown et al., 2014) was data-mined for genes potentially involved in metal transformation. Pathways relevant to metal transformation include 2 loci encoding arsenate reductase (UFO1_2328, UFO1_2536), subunits for a cytochrome c-dependent nitrate reductase (UFO1_1541, UFO1_1544), and several loci for flavin reductases (thought to be involved in chromium transformation). A predicted Co-Zn-Cd efflux system (UFO1_2233) is also present. Additionally, “P. lilae” UFO1 has a chrA locus for chromate transport (UFO1_3236), and a NiFe-hydrogenase (UFO1_2674) linked to a cytochrome b (UFO1_2673). Obviously, characterized metal-reducing Firmicutes, including strain UFO1, harbor a multiplicity of genes conferring metal detoxification ability, further extending the importance of these organisms in the biogeochemical processing of toxic metals.

Potential Ecophysiologic Role of “Pelosinus lilae” UFO1 in Subsurface Environments

“Pelosinus lilae” UFO1 represents a novel addition to a recently described and poorly characterized genus of fermentative bacteria. The most closely-related species, P. fermentans R7T, was similarly isolated from an Fe(III)-reducing enrichment; however, the enrichment was established with kaolin lenses originating from Plast, Russia (Shelobolina et al., 2007). This suggests that representatives of Pelosinus may be widespread in anoxic environmental systems. Kappler et al. (2004) demonstrated that fermenting bacteria were one of the largest populations of bacteria in freshwater lake sediments and had an important role in humic acid reduction; the results suggested that humic acid-mediated reduction of poorly soluble Fe(III) oxides is an.
TABLE 4 | Distinguishing features of strain UFO1 compared to the most closely related described species in the Class Negativicutes.

| Trait                  | 1          | 2          | 3       | 4          | 5       | 6       | 8       |
|------------------------|------------|------------|---------|------------|---------|---------|---------|
| G+C content of DNA (mol %) | 38.0       | 41.0       | ND      | 51.5       | 52.0–54.0 | 35.0   | 39.2   |
| Cell shape             | Slightly curved or straight rods | Straight rods | Straight rods | Straight rods | Curved rods | Rods    |
| Spore formation        | +          | +          | +       | +          | +       | −       | ND      |
| Motility               | +          | +          | +       | +          | +       | +*      | ND      |
| Cell size (µm)         | 0.2−0.7 x 1.5−4.7 | 0.6 x 2−6  | 0.5−0.7 x 2.2−12  | 0.6 x 6−60  | 0.5 x 3   | 0.5 x 2−10  | 1 x 2−7 |
| Temperature range (°C) | 22−37      | 4−36       | 19−35   | 19−40      | 20−45   | 10−42   | 10−42   |
| Temperature optimum (°C) | 37        | 22−30      | 30      | 30−33      | 25−30   | 37      |
| pH range               | 5.5−8      | 5.5−8      | 6.2−8.2 | 6.4−8.6    | ND      | 5−8.5   | 7.0−7.5 |
| pH optimum             | 7          | 7          | 7.8     | 7.8        | ND      | 6.5−7.5 | 7.0     |

Growth on:
- Fructose + + + + + − ND
- Fumarate + + + + ND − +
- Glucose + + + + ND − ND
- Glycerol + − + + − + +
- H2+CO2 − − − + − − −
- Lactate + + + + − ND − ND
- Malate + + + + ND − ND − ND
- Mannitol + + ND + − − −
- Pyruvate + + + + − ND − ND
- Succinate − − − − − − −

1-Strain UFO1 (this study); 2 Pelosinus fermentans R7T (Shelobolina et al., 2007); 3 Pelosinus propionicus TonP3 (Boga et al., 2007; Moe et al., 2012); 4 Acetonema longum APO-1T (Kane and Breznak, 1991); 5 Dendrosporobacter quercicolus DSM 1736T (Stankewich et al., 1971; Strompl et al., 2000); 6 Anaerosinus glycerini strain LGS 4T (Schauder and Schink, 1989; Strompl et al., 1999); 8 Pelosinus defluvii Shi-1T (Moe et al., 2012). *motility lost after 1 year of cultivation. ND, not determined.
important reductive pathway in anoxic natural environments. “P. lilae” UFO1 demonstrated the ability to reduce the humic acid analog, AQDS, in the presence of H2 and fermentable substrates; in addition, AQDS mediated the reduction of the insoluble Fe(III)-oxide, ferricydrate. The presence of AQDS also enhanced Cr(VI) removal from solution. The AQDS-AH2DS couple has a standard potential (E°) of −184 mV at pH 7 (Fultz and Durst, 1982; Wolf et al., 2009), which is well below that for the CrO42−−Cr(OH)2 couple (+480 mV at pH 7) (Takeno, 2005); therefore, the transfer of electrons from AH2DS to CrO42−− is thermodynamically favorable and suggests Cr(VI) removal in the presence of AQDS was likely due to reduction. These findings suggest that fermentative bacteria such as strain UFO1 may play a role in the reduction of humic acids, which may in turn facilitate the reduction of metallic contaminants in subsurface environments.

“Pelosinus lilae” UFO1 was isolated from pristine sediments beneath Oak Ridge National Laboratory, Oak Ridge, TN, United States. Organisms with high 16S rRNA gene sequence similarity to strain UFO1 have been detected in Fe(III)-reducing (Petrie et al., 2003) and U(VI)-reducing (Nyman et al., 2007) enrichments initiated with contaminated sediments, which may suggest the prevalence of Pelosinus species in the subsurface. Strain UFO1, and organisms with similar metabolic capabilities also offer potential for the removal of soluble U(VI) from cell suspensions (Ray et al., 2011). “P. lilae” UFO1 has been shown to reduce metals with or without an exogenous “electron shuttle” such as AQDS, and genome analysis predicts additional metal-transformation pathways that have not yet been verified. The transformation of an array of electron acceptors such as AQDS, Fe(III), Cr(VI) and U(VI) by UFO1 suggest a potentially important role for this and similar organisms in influencing the biogeochemistry of pristine and contaminated geologic media at the ORFRC.

REFERENCES
Amann, G., Stetter, K. O., Llobet-Brossa, E., Amann, R., and Anton, J. (2000). Direct proof for the presence and expression of two 5% different 16S rRNA genes in individual cells of Halosarcia marismortui. Extremophiles 4, 373–376. doi: 10.1007/s007920070007
Balch, W. E., and Wolfe, R. S. (1976). New approach to the cultivation of methanogenic bacteria: 2-mercaptoethanesulfonic acid (HS-CoM)-dependent growth of Methanobacterium ruminantium in a pressurized atmosphere. Appl. Environ. Microbiol. 32, 781–789.
Beller, H. R., Han, R., Karaouz, U., Lim, H., and Brodie, E. L. (2013). Genomic and physiological characterization of the chromate-reducing, aquifer-derived firmicute Pelosinus sp. strain HCF1. Appl. Environ. Microbiol. 79, 63–73. doi: 10.1128/AEM.02496-12
Benz, M., Schink, B., and Brune, A. (1998). Humic acid reduction by Propionibacterium freudenreichii and other fermenting bacteria. Appl. Environ. Microbiol. 64, 4507–4512.
Bochner, B. R. (1989). Sleuthing out bacterial identities. Nature 339, 157–158. doi: 10.1038/339157a0
Boga, H., Ji, R., Ludwig, W., and Brune, A. (2007). Sporotalea propionica gen. nov. sp. nov., a hydrogen-oxidizing, oxygen-reducing, propionigenic firmucute from the intestinal tract of a soil-feeding termite. Arch. Microbiol. 187, 15–27. doi: 10.1007/s00203-006-0168-7
Borch, T., Inskeep, W. P., Harwood, J. A., and Gerlach, R. (2005). Impact of ferricydrate and anthraquinone-2,6-disulfonate on the reductive transformation of 2,4,6-trinitrotoluene by a gram-positive fermenting bacterium. Environ. Sci. Technol. 39, 7126–7133. doi: 10.1021/es0504441

AUTHOR CONTRIBUTIONS
AR, SC, AN, JL, and DC conducted the experiments. AR, SC, VF, DC, and TM designed the experiments. All authors contributed to the preparation of the manuscript.

FUNDING
This work was supported by funding from the Inland Northwest Research Alliance (INRA) and Idaho National Laboratory (INL) under DOE Idaho Operations Office Contract DE-AC07-05ID14517. AR was supported by INRA and INL Subsurface Science Graduate Fellowship, TM and DC were supported by INRA Subsurface Science Initiative Grant Number ISU-004.

ACKNOWLEDGMENTS
We thank E. O’Leary-Jepson and M. Andrews of the ISU MRCF for DNA sequencing; P. Pinheiro and T. Trowbridge for help with electron microscopy; J. Taylor for ion chromatography; M. Thomas for assistance with phylogenetic analysis; A. Eddingsaas, L. Wendt, T. Gresham, D. Reed, and J. Barnes for technical assistance. We would also like to thank D. Watson for providing core samples from ORFRC. XPS was performed at the Image and Chemical Analysis Laboratory, Department of Physics, Montana State University, Bozeman, MT, United States.
Lovley, D. R., Phillips, E. J. P., and Lonergan, D. J. (1989). Hydrogen and formate oxidation coupled to dissimilatory reduction of iron or manganese by *Alteromonas putrefaciens*. Appl. Environ. Microbiol. 55, 700–706.

Lovley, D. R., Stolz, J. F., Nord, G. L., and Phillips, E. J. P. (1987). Anaerobic production of magnetite by a dissimilatory iron-reducing microorganism. Nature 330, 252–254. doi: 10.1038/330252a0

Magnuson, T. S., Neal, A. L., and Geesey, G. G. (2004). Combining in situ reverse transcriptase polymerase chain reaction, optical microscopy, and X-ray photoelectron spectroscopy to investigate mineral surface-associated microbial activities. *Microb. Ecol.* 48, 578–588. doi: 10.1007/s00248-004-0253-x

McIntyre, N. S., and Zetaurki, D. G. (1977). X-ray photoelectron spectroscopic studies of iron oxides. *Anal. Chem.* 49, 1521–1529. doi: 10.1021/ac50019a016

Mesbah, M., Premachandran, U., and Whitman, W. B. (1992). Precise measurement of the G+C content of deoxyribonucleic acid by high-performance liquid-chromatography. *Int. J. Syst. Bacteriol.* 39, 159–167. doi: 10.1099/00270713-39-2-159

Miller, T. L., and Wolin, M. J. (1974). A serum bottle modification of the Hungate technique for cultivating obligate anaerobes. *Appl. Environ. Microbiol.* 27, 985–987.

Moe, M. W., Stemberg, R. E., Rao, J. Y., Bowman, K. S., Nobre, N. F., da Costa, M. S., et al. (2012). *Pelosinus deflavi* sp. nov., isolated from chlorinated solvent-contaminated groundwater, emended description of the genus *Pelosinus* and transfer of *Spotarolera propionica* to *Pelosinus propionicus* comb. nov. *Int. J. Syst. Microbiol.* 62, 1356–1376. doi: 10.1099/ijsym.0.033753-0

Moore, L. V., Bourne, D. M., and Moore, W. E. (1994). Comparative distribution and taxonomic value of cellular fatty acids in thirty-three genera of anaerobic gram-negative bacilli. *Int. J. Syst. Bacteriol.* 44, 338–347. doi: 10.1099/00270713-44-2-338

Morandi, A., Zhaxybayeva, O., Gogarten, J. P., and Graf, J. (2005). Evolutionary and diagnostic implications of intragenomic heterogeneity in the 16S rRNA gene in *Aeronomas* strains. *J. Bacteriol.* 187, 6561–6564. doi: 10.1128/JB.187.16.6561-6564.2005

Mylvaganam, S., and Dennis, P. P. (1992). Sequence heterogeneity between the two genes encoding 16S rRNA from the halophilic archaeabacterium *Halocarcula marismortui*. *Genetics* 130, 399–410.

NABIR (2003). *Biomethylation of Metals and Radionuclides: What it is and How it Works*. Available at: https://escholarship.org/uc/item/7md2589q

Neal, A. L., Clough, L. K., Perkins, T. D., Little, B. J., and Magnuson, T. S. (2004). In situ measurement of Fe(III) reduction activity of *Geobacter pelophilus* by simultaneous in situ RT-PCR and XPS analysis. *FEBS Microbiol. Ecol.* 49, 163–169. doi: 10.1078/febsm.2004.03.014

Newsome, L., Morris, K., Trivedi, D., Bewsher, A., and Lloyd, J. R. (2015). Biostimulation by glycolate phosphate to precipitate recalcitrant uranium(VI) phosphate. *Environ. Sci. Technol.* 49, 11070–11078. doi: 10.1021/acs.est.5b02042

Neal, A. L., Techkarnjanaruk, S., Dohnalkova, A., McCready, D., Peyton, B. M., and Geesey, G. G. (2001). Iron sulfides and sulfur species produced at hematite surfaces in the presence of sulfate-reducing bacteria. *Geochim. Cosmochim. Acta* 65, 223–235. doi: 10.1016/S0016-7037(00)00537-8

Nevin, K. P., and Lovley, D. R. (2000). Potential for nonenzymatic reduction of Fe(III) via electron shuttling in subsurface sediments. *Environ. Sci. Technol.* 34, 2472–2478. doi: 10.1021/es991181b

Nubel, U., Engelen, B., Feske, A., Snaider, J., Wieshuber, A., and Amann, R. I., et al. (1996). Sequence heterogeneities of genes encoding 16S rRNAs in *Parvibacillus polymyxa* detected by temperature gradient gel electrophoresis. *J. Bacteriol.* 178, 5636–5643. doi: 10.1128/JB.178.19.5636-5643.1996

Nyman, J., Gentile, M., and Criddle, C. (2007). Sulfate requirement for the growth of U(VI)-reducing bacteria in an ethanol-fed enrichment. *Bioremediat. J.* 11, 21–32. doi: 10.1080/10991811003316790

Parte, A. C. (2014). LPSN—list of prokaryotic names with standing in nomenclature. *Nucleic Acids Res.* 40, D613–D616. doi: 10.1093/nar/gkt1111

Petrie, L., North, N. N., Dollhopf, S. L., Balkwill, D. L., and Kostka, J. E. (2003). Enumeration and characterization of iron(III)-reducing microbial communities from acidic subsurface sediments contaminated with uranium(VI). *Appl. Environ. Microbiol.* 69, 7467–7479. doi: 10.1128/AEM.69.12.7467-7479.2003

Pratt, A. R., Mui, I. J., and Nesbitt, H. W. (1994). X-ray photoelectron and Auger electron spectroscopic studies of pyrrhotite and mechanism of air
oxidation. Geochim. Cosmochim. Acta 58, 827–841. doi: 10.1016/0016-7037(94)90508-8
Quast, C., Pruesse, E., Yilmaz, P., Gerken, J., Schweer, T., Yarza, P., et al. (2013). The SILVA ribosomal RNA gene database project: improved data processing and web-based tools. Nucleic Acids Res. 41, D590–D596. doi: 10.1093/nar/gks1219
Ray, A. E., Barg, J. R., Sivaswamy, V., Dohnalkova, A. C., Fujita, Y., Peyton, B. M., et al. (2011). Evidence for multiple modes of uranium immobilization by an anaerobic bacterium. Geochim. Cosmochim. Acta 75, 2684–2695. doi: 10.1016/j.gca.2011.02.040
Ray, A. E., Connon, S. A., Sheridan, P. P., Gilbreath, J., Shields, M., Newby, D. T., et al. (2010). Intragenomic heterogeneity of the 16S rRNA gene in strain UF01 caused by a 100 bp insertion in helix 6. FEMS Microbiol. Ecol. 72, 343–353. doi: 10.1111/j.1574-6941.2010.00886.x
Reasoner, D. J., and Geldreich, E. E. (1985). A new medium for the enumeration and subculture of bacteria from potable water. Appl. Environ. Microbiol. 49, 1–7.
Riley, R. G., Zachara, J. M., and Wobber, F. J. (1992). Chemical Contaminants on Subsurface Kaolin Lenses. US-DOE.
Shirley, D. A. (1972). High-resolution X-ray photoemission spectrum of the valence bands of gold. Phys. Rev. B, 5, 4709–4714. doi: 10.1103/PhysRevB.5.4709
Shelobolina, E. S., Nevin, K. P., Blakeney-Hayward, J. D., Johnsen, C. V., Plaa, T. W., Krader, P., et al. (2007). Geobacter pickeringii sp. nov., Geobacter argillicus sp. nov. and Pelosinus fermentans gen. nov., sp. nov., isolated from subsurface kaolin lenses. Int. J. Syst. Evol. Microbiol. 57, 126–135. doi: 10.1099/ijs.0.64221-0
Shen, H., and Wang, Y. T. (1993). Characterization of enzymatic reduction of hexavalent chromium by Escherichia coli ATCC 33456. Appl. Environ. Microbiol. 59, 3771–3777.
Shirley, D. A. (1972). High-resolution X-ray photoemission spectrum of the valence bands of gold. Phys. Rev. B, 5, 4709–4714. doi: 10.1103/PhysRevB.5.4709
Stankewich, J. P., Cosenza, B. J., and Shigo, A. L. (1971).
Strompl, C., Tindall, B. J., Jarvis, G. N., Lunsford, H., Moore, E. R. B., and Hippe, H. (1999). A re-evaluation of the taxonomy of the genus Anaerovibrio, with the reclassification of Anaerovibrio glycerini as Anaerosomas glycerini gen. nov., comb. nov., and Anaerovibrio burkinabensis as Anaeroarcus burkinabensis [corr.]. gen. nov., comb. nov. Int. J. Syst. Bacteriol. 49, 1861–1872. doi: 10.1099/00277713-49-4-1861
Strompl, C., Tindall, B. J., Lunsford, H., Wong, T. Y., Moore, E. R. B., and Hippe, H. (2000). Reclassification of Clostridium quercicolus as Dendrosporobacter quercicolus gen. nov., comb. nov. Int. J. Syst. Evol. Microbiol. 50, 101–106. doi: 10.1099/00277713-50-1-101
Strunk, O., Ludwig, W., Gross, O., Reichel, B., May, M., Hermann, S., et al. (1996). ARB—a Software Environment for Sequence Data, 2.5b. Munich: Technical University of Munich.
Swift, P. (1982). Adventitious carbon–the panacea for energy referencing? Surf. Interface Anal. 4, 47–51. doi: 10.1002/sia.747040204
Tabak, H., Lens, P., van Hullebusch, E., and Dejonghe, W. (2005). Developments in bioremediation of soils and sediments polluted with metals and radionuclides – 1. Microbial processes and mechanisms affecting bioremediation of metal contamination and influencing metal toxicity and transport. Rev. Environ. Sci. Biotechnol. 4, 115–156. doi: 10.1007/s11157-005-1694-1
Takino, N. (2005). Atlas of Eh-pH diagrams. Intercomparison of thermodynamic databases. Geol. Surv. Japan Open File Rep. 419:285.
Thorsersen, M. P., Lancaster, W. A., Rajeev, L., Ge, X., Vaccaro, B. J., Poole, F. L., et al. (2017). A highly expressed high-molecular-weight S-layer complex of Pelosinus sp. strain U01 binds uranium. Appl. Environ. Microbiol. 83:e03044-16. doi: 10.1128/AEM.03044-16
Tourouva, T. P., Kuznetsova, B. B., Novikova, E. V., Poltaraus, A. B., and Nazina, T. N. (2001). Heterogeneity of the nucleotide sequences of the 16S rRNA genes of the type strain of Desulfitonocatulum kuznetsovii. Microbiology 70, 678–684. doi: 10.1023/A:1013135831669
USDOE (1997). Linking Legacies Report: Connecting the Cold War Nuclear Weapons Production Processes to Their Environmental Consequences. Washington, DC: U.S. Department of Energy.
Velasquez, L., and Dussan, J. (2009). Biosorption and bioaccumulation of heavy metals on dead and living biomass of Bacillus sphaericus. J. Hazard. Mater. 167, 713–716. doi: 10.1016/j.jhazmat.2009.01.044
Wayne, L. G., Brenner, D. J., Colwell, R. R., Grimont, P. A. D., Kandler, O., Krichevsky, M. I., et al. (1987). Report of the ad hoc committee on reconciliation of approaches to bacterial systematics. Int. J. Syst. Bacteriol. 37, 463–464. doi: 10.1099/00207713-37-4-463
White, C., Sayer, J. A., and Gadd, G. M. (1997). Microbial solubilization and immobilization of toxic metals: key biogeochemical processes for treatment of contamination. FEMS Microbiol. Rev. 20, 503–516. doi: 10.1111/j.1574-6976.1997.tb00333.x
Wolf, M., Kappler, A., Jiang, J., and Meckenstock, R. U. (2009). Effects of humic substances and quinones at low concentrations on ferricydrate reduction by Geobacter metallireducens. Environ. Sci. Technol. 43, 5679–5685. doi: 10.1021/es803647r
Woo, P. C. Y., Teng, J. L. L., Leung, K. W., Lau, S. K. P., Woo, G. K. S., Wong, A. C. Y., et al. (2005). Anaerospira hongkongensis gen. nov. sp. nov., a novel genus and species with ribosomal RNA operon heterogeneity isolated from an intravenous drug abuser with pseudobacteraemia. Microbiol. Immunol. 49, 31–39. doi: 10.1111/j.1574-6976.2005.tb06367.x

Conflict of Interest Statement: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

Copyright © 2018 Ray, Connon, Neal, Fujita, Cummings, Ingram and Magnuson. This is an open-access article distributed under the terms of the Creative Commons Attribution License (CC BY). The use, distribution or reproduction in other forums is permitted, provided the original author(s) and the copyright owner(s) are credited and that the original publication in this journal is cited, in accordance with accepted academic practice. No use, distribution or reproduction is permitted which does not comply with these terms.