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Investigation of the structures of sulfur on Mo(100) by scanning tunneling microscopy

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The structures of the four ordered overlayers of sulfur on the Mo(100) surface were investigated by scanning tunneling microscopy (STM). In order of increasing sulfur coverage, the overlayers have \( c(2 \times 2) \), \( c(4 \times 2) \), and \( p(2 \times 1) \) low energy electron diffraction (LEED) patterns. Only the \( c(2 \times 2) \) overlayer geometry has been determined from LEED \( I(V) \) analysis. An examination of point and line defects and domain boundaries in STM images provides information beyond that known from LEED on the required symmetries of the ordered overlayers. Several proposed structures were shown to be incompatible with these symmetries and therefore discarded. Only one model of the \( c(4 \times 2) \) structure was found to be consistent with the symmetry of the STM images. This structure contains sulfur atoms occupying exclusively fourfold hollow sites. Unexpectedly, STM images of the surface with the \( p(2 \times 1) \) LEED pattern did not have \( p(2 \times 1) \) symmetry, but consisted of small domains of the \( c(4 \times 2) \) structure.

I. INTRODUCTION

Molybdenum single crystals surfaces are of interest as a model for industrial molybdenum sulfide catalysts. A number of studies have been published on the sulfided Mo(100) surface. These studies have determined that sulfur forms several ordered structures as a function of coverage. While early reports conflicted over the order and number of distinct low energy electron diffraction (LEED) patterns, more recent results agree on four LEED patterns and the sulfur coverages at which they appear. However, except for the \( c(2 \times 2) \) ordered overlayer, the structures of the unit cells which produce these LEED patterns are unknown. Several models have been proposed for each of the other overlayers. Scanning tunneling microscopy (STM), by providing information complimentary to the LEED results, should be helpful in determining which of the proposed models is correct.

The most recently published studies of S on Mo(100) agree on four ordered structures. At coverages of 0.5, 0.67, 0.75, and 1 monolayer of sulfur the reported LEED patterns have \( c(2 \times 2) \), \( c(4 \times 2) \), \( c(2 \times 1) \) symmetry. These coverages correspond to one, two, three, and two atoms per unit cell, respectively. A LEED calculation performed on the \( c(2 \times 2) \) structure determined that the sulfur atoms reside at fourfold hollow sites. A later dynamical LEED and a tensor LEED calculation determined the precise position of the sulfur adatoms and the displacements of the substrate Mo surface atoms in this structure. Dynamical LEED calculations have also been attempted for the \( p(2 \times 1) \) overlayer, but an adequate fit between the calculation and experimental \( I(V) \) curves has not been obtained. A previous STM study of this structure in air favored a model with all sulfur atoms occupying an asymmetric site between hollow and bridge. However, the symmetries of this structure would cause the extinction of some spots in the LEED pattern, which is not observed experimentally. A total energy calculation favored a model with half the sulfur atoms in hollow sites and the other half in bridge sites.

In our attempt to determine the structure of these overlayers we considered models with sulfur atoms in bridge and hollow sites. Unfortunately, we were not able to directly determine which of these structures is correct. Although the unit cell periodicity is resolved by the STM, the constituent atoms are not resolved individually except in the simple \( c(2 \times 2) \) structure. However, proposed structures could be eliminated from consideration by showing them to be incompatible with observations of the symmetry of the overlayer and of point defects and domain boundaries.

II. EXPERIMENT

Sample preparation and STM imaging were performed inside a standard surface science ultra high vacuum (UHV) chamber with a base pressure of \( 5 \times 10^{-10} \) Torr. Besides the STM, the chamber was equipped with Auger electron spectroscopy (AES), LEED, and \( Ar^{+} \) sputtering. The design of the STM is described elsewhere. Mechanically cut Pt/Rh (80%/20%) tips were used. The single crystal sample could be separately heated with an electron
beam heater and cooled by placing the sample holder in contact with a copper block cooled by liquid nitrogen. STM imaging took place at room temperature.

The single crystal sample was approximately 1 cm diam and < 1 mm thick. It was oriented to within < 1.5° of the [100] direction and polished using standard metallurgical techniques. Carbon and other impurities were removed by sputtering while heating to approximately 600 K. The carbon remaining after this treatment was removed by repeated heating in oxygen (5×10⁻⁷ Torr at 1600 K). Remaining oxygen was removed by heating the crystals in UHV above 1900 K. Careful cleaning of the crystal was important to insure that only sulfur structures were imaged with the STM. Carbon and oxygen, the most common impurities on the surface, form ordered structures with some of the same LEED patterns as sulfur. Unless the bulk is depleted of these impurities to a sufficient degree they will segregate back to the surface while the crystal is cooled.

Sulfur was deposited on the clean surface by decomposition of H₂S gas and with a solid-state electrochemical source (Ag/Ag₁/Ag₂S). While the lower coverage structures could be produced easily by H₂S exposure, the electrochemical source was necessary to cover the surface with enough sulfur to produce the highest coverage \( p(2 \times 1) \) overlayer. The crystal was heated during exposure to H₂S gas, while it was kept at room temperature during exposure to sulfur from the electrochemical source.

Prior to imaging of the sulfur overlayer, AES was used to determine sulfur coverage on the surface. The sample was annealed to produce an ordered LEED pattern. STM images of the surface were obtained using several different tips and preparations of each structure. Several hundred STM images of the overlayer structures were obtained. As discussed in a separate paper, sulfur adsorption was found to modify the structure of the atomic steps on the surface by causing the coalescence of steps and the consequent enlargements of the (100) terraces.

The surface was imaged in both constant current (topographic) and constant height mode. In most of the images the sample was biased negative relative to the tip. Changing the polarity of the bias was found to have no effect on the images. The tunneling current was set at between 1 and 3 nA and the bias voltage between 10 and 100 mV. Decreasing the gap resistance tended to increase the corrugation of the sulfur structures. Changing the bias had little effect on the image as long as the gap resistance was kept constant.

Thermally induced drifts of the sample during imaging significantly distorted the images, particularly the topographic mode images that require longer acquisition time. The correct shape and size of the images was determined from current mode images acquired quickly enough so that the drift had little effect on them. Using this information the images were replotted with the correct shape using a two-dimensional second-order fit of the lattice. All the images shown in this paper are drift-corrected topographic mode images. To make interpretation easier, the images are rotated so that the horizontal and vertical directions correspond to the [010] and [001] directions of the Mo substrate. They were also Fourier filtered to remove spatial frequency components too high to represent real information.

### III. RESULTS

#### A. c(2×2) structure

Images of the Mo(100) surface with half a monolayer of sulfur and a c(2×2) LEED pattern show ordered regions approximately 20–50 Å across separated by partially ordered regions. The boundaries between these regions run in the [010] directions of the substrate. The partially ordered regions contained maxima with approximately 0.8 Å corrugation which were common, but not always, spaced by twice the Mo lattice constant \( a \) of 3.15 Å. As the AES spectrum showed little carbon or oxygen contamination and the disordered regions were common on the surface, it is likely that these areas are due to a different structure of sulfur and not to an impurity. They probably consist of sulfur which has not ordered into the c(2×2) structure due to insufficient annealing or too low a sulfur coverage. These regions may produce the \( p(2 \times 2) \) and c(4×4) LEED patterns which have been reported by some authors at coverages well under half a monolayer.

An example of an image of an ordered region of this surface is shown in Fig. 1. The structure in the image is consistent with the observed LEED pattern. Rows of nearest-neighbor maxima run at 45° to the lattice directions of the substrate. Also the measured spacing between the maxima is within 10% of the 4.7 Å nearest-neighbor distance of a c(2×2) overlayer. The corrugation of the maxima in this image is 0.22 Å. Occasionally images of the c(2×2) structure were obtained which contained a smaller
FIG. 2. A model of the hollow site c(2×2)-S overlayer.

FIG. 3. A model of the Mo(100) surface showing the separation between high symmetry sites in two domains rotated by 90° relative to each other. A (1×1) lattice aligned on the fourfold symmetric top or hollow sites remains on the same points after a rotation through 90°. Conversely, a lattice aligned on bridge sites is displaced by a lattice vector of \([0, \pm a/2, \pm a/2]\) after a 90° rotation.

FIG. 4. A 25-Å square STM topographic image of the \([\frac{\sqrt{3}}{2}]\) structure of S on Mo(100). It consists of unresolved rows of sulfur atoms running diagonally in the image along the [011] direction. The lattice directions of the Mo substrate are the horizontal and vertical directions in the image. The corrugation of the rows is 0.27 Å. Bias = 100 mV, I = 1.0 nA.

FIG. 5. Three models of the \([\frac{\sqrt{3}}{2}]\) structure with sulfur atoms at bridge and hollow sites. (A) all sulfur atoms at hollow site. (B) A mixture of hollow and bridge sites. (C) All sulfur atoms at bridge sites.
Fig. 6. A 31×31 Å topographic STM image showing the boundary region between the two domains of the [111] structure (on the left, brighter rows) and the c(2×2) overlayer (on the right, dimmer spots) together. A (1×1) grid is drawn over the image and aligned to the c(2×2) overlayer. The cross marks on the grid are therefore positioned over fourfold hollow sites of the substrate. The grid shows that the rows of the [111] structure appear to be centered over bridge sites. Bias = 50 mV, I = 1.0 nA.

STM images of this structure could be explained by any of the three models. If we assume that, as in the c(2×2) structure, the sulfur atoms have positive corrugation, the rows of sulfur atoms in Figs. 5(A) and 5(C) could correspond to the rows in the image. However, the rows of bridge sites in model B could also produce the rows in the image. As the bridge atoms are higher above the surface than atoms at hollow sites one would expect the rows of bridge site atoms in model B to appear brighter in an STM image than the rows of hollow site atoms. The observed variations in the experimental shape of the rows can be explained in all cases as the effect of a particular tip structure and are not helpful in determining the correct unit cell structure.

Since the [111] structure coexisted on the surface with the c(2×2) structure it was possible to determine the relative position of the rows and the maxima of the c(2×2) overlayer in the STM images. Figure 6 shows an image of the boundary region between both structures. In this image the c(2×2) domain occupies the right-hand side and its maxima have a low average intensity. Two domains of the [111] structure rotated by 90° relative to each other appear as rows running diagonally up and down on the left-hand side. The intensity is higher in this region. A (1×1) grid is drawn over the image aligned to the maxima in the c(2×2) domain. If the hollow site c(2×2) model is correct, the top Mo substrate atoms are located at the center of the squares of the grid. The rows of the [111] structure appear to be centered over bridge sites of this lattice. This position is consistent with model B. If this is the correct model, the maxima at bridge sites appear elongated in one of the lattice directions. However, since we have not yet ascertained that sulfur atoms have positive corrugation in this structure, the result is not conclusive.

C. c(4×2) structure

At a coverage of approximately three quarters of a monolayer a sharp c(4×2) LEED pattern was obtained. STM images of the surface showed it was covered with ordered domains of approximately 100 Å in size. An example of an image of a single domain is shown in Fig. 7. The staggered vertical rows of maxima in the figure are separated by twice the Mo lattice vectors. The maxima along the rows are also separated by this distance. The symmetry of the maxima in this image is c(4×2), consistent with the LEED pattern. There are two perpendicular mirror planes in each unit cell of the STM image which run through the maxima in the unit cells. Six possible model structures, shown in Fig. 8, have these two mirror planes and contain the three atoms per unit cell required to have 0.75 monolayers coverage. In models A–C the perpendicular mirror planes cross at hollow sites while in models D–F they cross at bridge sites.

Half of these models can be discarded by observing the symmetry of domain boundaries in this structure. An image of the boundary between two domains of c(4×2) ordered sulfur rotated by 90° relative to each other [c(4×2) and c(2×2)] is shown in Fig. 9. A (1×1) lattice is plotted over this image and lined up with the maxima of one domain. In the rotated domain the lattice is also found to line up with the maxima. This indicates that the maxima must be located at fourfold (hollow or top) sites on the lattice.
FIG. 8. Models of the possible \( c(4 \times 2) \) unit cells which contain three sulfur atoms and have the two perpendicular mirror planes (continuous lines) apparent in the STM images. The mirror planes in models A–C cross over hollow sites. In models D–F they cross over bridge sites. Glide planes are shown as dashed lines.

If they were located over twofold (bridge) sites one domain would be shifted by \([0, \pm a/2, \pm a/2]\) relative to the other domain as shown in Fig. 2. The lack of this shift was observed on all of the images of domain boundaries. The maxima are at points in the image where the two perpendicular mirror planes cross. Therefore, the mirror planes cross over hollow sites, and models D–F must be discarded.

Observations of other domain boundaries support only one of the remaining models. Figure 10(a) shows an antiphase boundary in which one domain is shifted by one substrate lattice distance \( a \) relative to the other. Figure 10(b) is an image of a domain boundary region between two perpendicular domains in which the domains appear to "mix" into a \((2 \times 2)\) structure. The structure in this area can be explained as a periodic array of antiphase boundaries like that shown in Fig. 10(a). This region appears to have a fourfold symmetry, and extends away from both the \( c(4 \times 2) \) and \( c(2 \times 4) \) domains symmetrically. As two domains of models B and C contain two different orientations of bridge sites, these models may not extend symmetrically into a \((2 \times 2)\) region. A periodic array of antiphase boundaries in the structure of model A would have fourfold symmetry. Therefore model A, in which all sulfur atoms are located at hollow sites, is the most consistent with the STM images.

D. "\( p(2 \times 1) \)" structure

Producing a sample with a sharp \( p(2 \times 1) \) LEED pattern proved to be difficult. Since it is not possible to produce the required saturation coverage of sulfur by exposure...
to low pressures of H₂S, the electrochemical sulfur source was used to deposit sulfur on the surface. Annealing of the surface was necessary to obtain an ordered LEED pattern. Excessive annealing produced the c(4×2) pattern while insufficient annealing left the LEED pattern diffuse. Previous reports of this structure have stated that the LEED pattern was often streaked, at least at some sulfur coverages. In the present experiment the overlayer spots of the LEED pattern were streaked in the direction of the quarter order spots of the c(4×2) LEED pattern at some electron energies. Adjusting the electron energy changed the intensity of the streaking; at some energies it was nearly invisible.

Unlike the previous lower coverage structures, we were unable to obtain STM images of a surface structure consistent with the symmetry of the LEED pattern. This result is different from the previous STM study of this surface in air in which images of a surface with apparent p(2×1) symmetry were obtained. Because of the air environment the observed p(2×1) ordering may be due to some impurity which was absent in our UHV experiment.

A representative image of the surface with a p(2×1) LEED pattern is shown in Fig. 11. It consists of very small domains of c(4×2) ordered sulfur. This image is not actually inconsistent with the LEED pattern, but only with its apparent p(2×1) symmetry. The c(4×2) LEED pattern contains all the spots of the p(2×1) pattern and additional quarter order spots. The streaking we observed in the p(2×1) LEED pattern was in the direction of these spots. The lack of large domains on the surface would cause streaking of the higher-order spots as they correspond to longer range order on the surface. The higher coverage of sulfur on this surface relative to the ordered c(4×2) overlayer may be explained by the large number of antiphase defects present which have a local coverage of one monolayer. As other authors have also reported streaking of the p(2×1) overlayer LEED spots, it is possible that all reports of the p(2×1) LEED pattern were obtained from surfaces with small c(4×2) domains. It may also be that some impurity is required to induce p(2×1) ordering of the sulfur.

IV. CONCLUSION

By careful analysis of the symmetry of STM images, particularly those containing point defects and domain boundaries, the range of possible structures of sulfur on Mo(100) was narrowed. STM images of the c(2×2) structure are consistent with the results obtained from dynamic LEED calculations. Of a number of proposed models of the c(4×2) structure only one model was found to be consistent with the STM images. This model consists of sulfur atoms adsorbed at fourfold hollow sites. Real space imaging by the STM has shown that the p(2×1) LEED pattern does not correspond to a true p(2×1) ordering. The structure of the p(2×1) overlayer is still unknown.

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1M. E. Bussell and G. A. Somorjai, J. Catal. 106, 93 (1987).
2M. E. Bussell, A. J. Gellman, and G. A. Somorjai, Catal. Lett. 1, 195 (1988).
3M. Salmeron, G. A. Somorjai, and R. R. Chianelli, Surf. Sci. 127, 526 (1983).
4V. Maurice, L. Peralta, Y. Berthier, and J. Oudar, Surf. Sci. 148, 623 (1984).
5L. J. Clarke, Surf. Sci. 102, 331 (1981).
6F. Zaera, E. B. Kollin, and J. L. Gland, Surf. Sci. Lett. 166, L149 (1986).
7M. H. Farias, A. J. Gellman, G. A. Somorjai, R. R. Chianelli, and K. S. Liang, Surf. Sci. 140, 181 (1984).
8C. C. Knight and G. A. Somorjai, Surf. Sci. 240, 101 (1990).
9C. C. Knight and G. A. Somorjai, Surf. Sci. 272, 326 (1992).
10J. C. Dunphy, C. Knight, P. Sautet, D. F. Ogletree, G. A. Somorjai, and M. B. Salmeron, Surf. Sci. (in press).
11D. G. Kelley, R. F. Lit, M. A. Van Hove, and G. A. Somorjai, Surf. Sci. 224, 97 (1989).
12P. J. Rous, D. Jentz, D. G. Kelley, R. Q. Hwang, M. A. Van Hove, and G. A. Somorjai, The Structure of Surfaces III, Proceedings of the 3rd International Conference on the Structure of Surfaces (ICOSIS-III), edited by S. Y. Tong, M. A. Van Hove, K. Takayanagi, and X. D. Xie (Springer, Berlin, 1991), p. 432.
13D. Jentz and M. A. Van Hove (private communication).
14B. Marchon, P. Bernhardt, M. E. Bussell, G. A. Somorjai, M. Salmeron, and W. Siekhaus, Phys. Rev. Lett. 60, 1166 (1988).
15X. W. Wang and S. G. Louie, Surf. Sci. 226, 257 (1990).
16D. M. Zegliniski, D. F. Ogletree, T. P. Beebe, Jr., R. Q. Hwang, G. A. Somorjai, and M. Salmeron, Rev. Sci. Instrum. 61, 3769 (1990).
17. C. Wagner, J. Chem. Phys. 21, 1819 (1953).
18. A. J. Gellman, J. C. Dunphy, and M. Salmeron, Langmuir 8, 534 (1992).
19. V. Maurice and P. Marcus, Surf. Sci. 262, L59 (1992).
20. J. C. Dunphy, D. F. Ogletree, M. Salmeron, P. Sautet, and M.-L. Bocquet, Ultramicros. 42-44, 490 (1992).
21. S. Rousset, S. Gauthier, O. Siboulet, W. Sachs, M. Belin, and J. Klein, J. Vac. Sci. Technol. A 8, 302 (1990).
22. J. M. MacLauren, J. B. Pendry, P. J. Rous, D. K. Saldin, G. A. Somorjai, M. A. Van Hove, and D. D. Vvedensky, A Handbook of Surface Structures (Reidel, Dordrecht, 1987).