Polyamines in Chloride-stressed *Citrus* Plants: Alleviation of Stress by Nitrate Supplementation via Irrigation Water

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Abstract. A study was conducted to elucidate the effects of chloride in the irrigation water on growth and development of two citrus rootstocks. ‘Cleopatra’ mandarin (*Citrus reshni* Hort. ex Tan) is salt tolerant and ‘Troyer’ citrange (*Poncirus ×Citrus sinensis*) is salt sensitive. Increasing chloride from 2 to 48 mm in the irrigation water resulted in increased leaf chloride levels, more severe damage of the leaves, and reduced branch growth. High chloride in the irrigation water also caused increased putrescine (PUT) and decreased spermine (SPM) contents of the leaves. These effects were slight in ‘Cleopatra’ but highly apparent in ‘Troyer’. The symptoms caused by high chloride were associated with high PUT and low SPM levels in the leaves. PUT may be involved in the development of chloride toxic symptoms, and SPM may protect or have no effect on chloride plant injury. The leaf polyamine profiles of ‘Troyer’ and ‘Cleopatra’ under nonstress chloride conditions were different. In ‘Troyer’ leaves, PUT level was 9-fold higher than in ‘Cleopatra’; in ‘Cleopatra’ leaves, SPM level was 25-fold higher than in ‘Troyer’. Nitrate supplement to saline water reduced chloride accumulation in the leaves and reduced the increase in PUT. The possible connection between ethylene production and PUT and SPM levels in the leaves of stressed plants is discussed.

Citrus trees are known to be sensitive to salt (Chapman, 1968; Syvertsen et al., 1993; Zekri, 1993). Salinity in water for agricultural use is expressed in terms of chloride content. Irrigation water in Israel generally contains 2 to 16 mm chloride, but can go up to 30 mm (personal communication). Fruit trees are specifically sensitive to chloride taken up by the roots and accumulating in the leaves (Bernstein, 1981). Large amounts of chloride have been found in leaves and fruit juice of citrus trees irrigated with water high in chloride (Syvertsen et al., 1993). The degree of chloride damage to citrus trees varies with rootstock characteristics (Gallasch and Galton, 1989; Zekri, 1993).

In field surveys, chloride uptake and toxicity have been found to be influenced by nitrate. Haas (1928) reported chloride uptake by avocado trees to be higher at low concentrations of nitrate in the soil. Chapman and Liebig (1940) found that 20 mm chloride in the water only produced leaf toxic symptoms in citrus trees under nitrate-deficient conditions. The content of chloride in the leaves was 2.78% to 3.90% of the dry matter in nitrate-deficient plants and 0.53% in plants with ample nitrate supply. A competition between chloride and nitrate uptake has been shown in tomatoes (Kafkafi et al., 1982), melon, lettuce (Feigin, 1985), wheat (Soliman et al., 1994), strawberry (Awang and Atherton, 1994), and several other plant species (Kafkafi, 1984). Bar et al. (1992) showed chloride uptake and toxicity decrease in avocado seedlings of sensitive and tolerant rootstocks by increased nitrate concentration in the nutrient solution.

Stress has been shown to alter the polyamine titer in various plants (Smith, 1990). The polyamines spermidine (SPD) and spermine (SPM) and the diamine putrescine (PUT) are ubiquitous in living organisms and are associated with a variety of physiological plant processes (Apelbaum, 1990; Goren et al., 1982; Smith, 1990). PUT, and to a lesser extent SPD and SPM, have been shown to accumulate in higher plants in response to various environmental stresses (Altman, 1989). PUT accumulation has been reported in response to K and Mg deficiencies (Smith, 1984), ammonium nutrition (Corey and Barker, 1989; Feng and Barker, 1993), NaCl stress (Altman, 1989), acid stress (Young and Galston, 1983), heat stress (Das et al., 1987), and osmotic stress (Flores and Galston, 1982b).

Most studies on salinity effects deal with NaCl stress, and the effects of sodium and chloride ions are not separated. Some attempts have been made in citrus to distinguish between effects of sodium and chloride on photosynthesis, leaf gas exchange, growth rate, and defoliation (Banuls and Primo Millo, 1992; Lloyd et al., 1987, 1989, 1990; Walker et al., 1982). The aim of the present study was to test the effects of chloride on plant growth, leaf mineral composition, free polyamine titer, and toxic symptoms in salt-sensitive and salt-tolerant citrus plants under various nitrate regimes.

Materials and Methods

One-year-old plants of the salt-tolerant ‘Cleopatra’ mandarin (*Citrus reshni*) rootstock and the salt-sensitive ‘Troyer’ citrange (*Poncirus ×Citrus sinensis*) rootstock were grown in a greenhouse in 8-liter containers filled with sandy soil. The minimum temperatures during the experiment varied from 18 to 23°C and the maximum temperatures varied from 27 to 33°C. The relative humidity ranged from 40% to 50% to 75% to 95%. The field
capacity of the soil was determined and the plants were irrigated with 40% excess of the water required to reach field capacity. Irrigation occurred at 25% to 30% water loss through evapotranspiration. The plants were irrigated with nutrient solutions containing 2, 16, or 48 mM chloride during the 3-month experimental period. The chloride nutrient solutions were also supplemented with 2, 8, or 16 mM nitrate, resulting in a total of nine treatments, with three plants in each treatment. The relevant chloride and nitrate concentrations were obtained with chloride and nitrate salts of Mg, Ca, and K at a charge equivalent cation ratio of 1:5:3, respectively. The only sodium in the nutrient solutions were chemical impurities of the salts and traces of sodium in the deionized water. Ten percent of the nitrogen in the nutrient solutions was supplied as ammonium to prevent high pH near the roots, which may lead to P and Fe deficiencies (Bar, 1989; Bar and Kafkafi, 1992). All solutions also included 0.6 mM monopotassium phosphate, 0.2 mM magnesium sulfate, 1 mM calcium sulfate, and a commercial mix of chelated micronutrients (Korateen, Fertilizers and Chemicals Ltd., Haifa, Israel).

The treatments were arranged in a completely randomized experimental design. Eight branches of similar size from each plant were marked at the beginning of the experiment to follow growth rate and toxic symptoms. The growth rate of each branch was obtained by subtracting the branch length at the beginning of the experimental period from the length at the end of the experiment. Shoot damage was evaluated according to the following scale: 0 = no damage; 1 = scorched leaf tips; 2 = scorched leaf tips and margins; 3 = one-third of the leaf scorched; 4 = most of the leaf area scorched; 5 = severe leaf scorching and branch tips burning.

Exogenous polyamines. To study the effects of exogenously applied polyamines, uniform mature leaves of both rootstocks were cut below the laminar abscission zone. The cut surfaces of the excised leaves were dipped for 10 to 14 days in water containing 0.0, 0.01, 0.1, 1.0, or 10.0 mM PUT, SPD, or SPM, or a combination of 10 mM PUT and 10 mM SPD or 10 mM SPM. There were five leaves in each treatment. The solutions were changed daily. The leaves were placed in a growth chamber at 28 ± 1.5°C under a 14-h photoperiod at about 400 µE·m⁻²·s⁻¹ and examined daily for damage.

Leaf mineral analysis. At the end of the 3 months, all the mature leaves were collected and oven-dried at 70°C for 48 h and ground. Nitrogen, potassium, and magnesium were analyzed following wet digestion with H₂SO₄ and H₂O₂. Nitrogen was determined by colorimetry using Nessler’s reagent. Magnesium and potassium were determined by inductively coupled plasma emission spectroscopy. Chloride was water extracted and analyzed by chloridometer.

Free polyamine determination. Leaves from the middle of the branches were collected from the treated plants at the end of the experiment. The leaves were ground with cold 5% HClO₄ (E. Merck, Darmstadt, Germany) using a prechilled mortar and pestle at an extraction ratio of 100 mg fresh weight/ml HClO₄. The homogenate was kept in an ice bath for 60 min, vortexed, and centrifuged for 20 min at 27,700 g at 4°C. The precipitate was dissolved in HClO₄, kept for 5 min in the ice bath, vortexed, and centrifuged. The first and second supernatant fractions were collected and stored at −20°C for free polyamine analysis (Flores and Galston, 1982a). Plant extracts and authentic standards of PUT, SPD, and SPM (Sigma Co., St. Louis) were benzoylated following the procedure described by Flores and Galston (1982a). Two milliliters of 2
n NaOH was added to 1 ml of the HClO₄ extract or to 50 µl of the polyamine standards and vortexed. Eight microliters of benzoyl chloride (BDH Chemicals Ltd., Poole, England) was added and the mixture and incubated for 20 min at room temperature, then 2 ml of saturated NaCl was added and the mixture was vortexed. Benzoyl-polyamines were extracted with 3 ml diethyl ether (Frutarom Ltd., Haifa, Israel) and centrifuged at 2000×g for 5 min, and 1.5 ml of the ether phase was collected, evaporated to dryness under nitrogen stream, and redissolved in 150 µl methanol (J.T. Baker Inc., Phillipsburg, N.J.). The benzoylated samples were stored at −20°C. The polyamine levels were determined by a liquid chromatograph (Pharmacia LKB, Uppsala, Sweden) equipped with an ultraviolet detector (Micromeritics, Goshen Springs, Ga.) at 254 nm. The solvent system consisted of isocratic 50% acetonitrile (Frutarom Ltd., Haifa, Israel) and 50% water at a flow rate of 0.7 ml·min⁻¹ (Flores and Galston, 1982a). This solvent system was found to produce the best separation of polyamine peaks. The benzoylated samples were injected into a fixed-volume 20-µl loop and eluted at room temperature through a 4.6 × 250-mm, 5-µm particle size C-18 reverse-phase column (Rainin Microsorb, Emeryville, Calif.). A computerized integration program (Barspec Inc., Mansfield, Ohio) was used to quantify polyamines by comparison with the polyamine standards.

Statistical analysis. Significant means were separated by Duncan’s multiple range test.

Results

Leaf chloride and N content, branch growth, and leaf scorching. Increasing concentration of chloride in the irrigation water used for 90 days resulted in higher leaf chloride content of the ‘Cleopatra’ and ‘Troyer’ plants. Leaf chloride content increased more rapidly for ‘Troyer’ than ‘Cleopatra’ plants. Leaf chloride level of ‘Cleopatra’ plants irrigated with 2 mm chloride was 0.22%, increased to 0.39% at 16 mm, and showed a 9-fold increase to 1.93% when the plants were irrigated with water containing 48 mm chloride (Fig. 1). When the more sensitive ‘Troyer’ plants were irrigated with what can be considered high-quality water (2 mm chloride), a relatively high leaf chloride level was recorded (1.02%). In the leaves of ‘Troyer’ plants irrigated with 16 mm chloride, the chloride content was 2.58%—7-fold higher than in the leaves of ‘Cleopatra’ plants irrigated with the same nutrient solution.

High concentrations of chloride in the water inhibited branch growth (Fig. 1). The inhibition was more pronounced in ‘Troyer’ than in ‘Cleopatra’. When the chloride concentration in the water increased from 2 to 16 mm, the growth of ‘Cleopatra’ branches was suppressed from 45.4 to 32.3 cm (29%), whereas the growth of ‘Troyer’ branches decreased from 43.5 to 20.1 cm (54%) (Fig. 1). Increasing the chloride concentration from 16 to 48 mm did not have any additional effect on either rootstock.

Increasing the chloride concentration in the irrigation water did not affect N level in the leaves of either rootstock. Adding a high concentration of nitrate to the water (16 mm) increased N level in ‘Troyer’ leaves by 40% (from 2.4% to 3.37%) and in ‘Cleopatra’ leaves by 52% (from 2.08% to 3.16%) above that recorded at 2 mm nitrate.

Increased damage was observed in leaves and branches of ‘Troyer’ in response to increased water chloride concentration (Fig. 1). In addition to the leaf damage observed in ‘Troyer’, the branch tips were scorched at high chloride concentrations. ‘Cleopatra’ leaves showed no damage with increasing chloride concentration, except for slight yellowing at 48 mm chloride.

Effect of chloride concentration in the irrigation water on leaf polyamine levels. A marked difference in leaf polyamine profiles was found between ‘Troyer’ and ‘Cleopatra’ irrigated with water containing 2 mm chloride, considered nonstressful conditions (Fig. 2). The content of PUT in ‘Troyer’ leaves was 9-fold higher than in ‘Cleopatra’ (48.4 vs. 5.1 nmol·g⁻¹ fresh weight), and the level of SPD in ‘Cleopatra’ leaves was 3-fold higher than in ‘Troyer’ (1428 vs. 512 nmol·g⁻¹ fresh weight). The most pronounced difference was found in SPM, which was 25-fold higher in ‘Cleopatra’ leaves than in ‘Troyer’ (1409 vs. 55.5 nmol·g⁻¹ fresh weight). When the plants were irrigated with water containing increasing chloride concentrations from 2 to 48 mm, the level of PUT in the leaves of ‘Cleopatra’ increased from 5.1 to 13.8 nmol·g⁻¹ fresh weight and in ‘Troyer’ from 48.4 to 232.5 (Fig. 3), which is 17-fold higher than that in the ‘Cleopatra’. However, the levels of SPD and SPM remained unchanged in ‘Cleopatra’ leaves, whereas in ‘Troyer’ leaves the SPD level decreased from 512 to 406 nmol·g⁻¹ fresh weight and that of SPM from 55.5 to 31.1 (Fig. 3).

Polyamine treatments and leaf damage. When the cut surfaces of excised leaves were dipped in 0.01 or 0.1 mm solutions of PUT, SPD, or SPM, the leaves did not show any signs of damage for 14 days. Increasing the concentration of the polyamines to 1.0 mm resulted in leaf damage only with the PUT treatment, whereas leaves treated with up to 10 mm SPD or SPM for 14 days exhibited no damage (data not shown). ‘Troyer’ leaves were more sensitive than ‘Cleopatra’ leaves to the 1.0 mm PUT treatment, showing first signs of damage after 3 days; by day 5 a large portion of the leaves showed scorching damage and leaflet abscission. ‘Cleopatra’ leaves treated with 1.0 mm PUT did not show any damage for 10 days of treatment, after which they began to abscise. Treatment with 10 mm PUT resulted in more severe toxic symptoms in both rootstocks. Leaves treated with 10 mm SPD or SPM did not show any damage. Treatments with a combination of PUT and SPD or...
SPM did not affect the damage caused by PUT alone. Analysis of the treated leaves showed increased polyamine levels, which confirmed that polyamines had been taken up by the leaf tissue.

Nitrate supplementation of the irrigation water. Adding nitrate to chloride-containing irrigation water decreased leaf chloride content of ‘Troyer’ and ‘Cleopatra’ plants (Fig. 4). In ‘Cleopatra’, the reduction in chloride was more noticeable when the nitrate was added to water containing the higher chloride concentrations (16 and 48 mM). In the case of ‘Troyer’, the reduced chloride content was already noticeable at the lower chloride concentration. Nevertheless, chloride levels in leaves of plants receiving the same treatment were always higher in ‘Troyer’ than in ‘Cleopatra’.

A reduction in the severity of damage to ‘Troyer’ leaves and branches was recorded upon adding nitrate to irrigation water containing 16 or 48 mM chloride (Fig. 5). However, signs of chlorosis were apparent in ‘Troyer’ leaves of plants irrigated with water containing a high concentration of nitrate (16 mM). Adding nitrate to the irrigation water enhanced branch growth in ‘Cleopatra’ when 8 mM nitrate was added to water containing 16 mM chloride (Fig. 5). However, the growth inhibition of ‘Troyer’ branches inflicted by chloride was markedly reduced by the adding 8 mM nitrate to water containing 2, 16, or 48 mM chloride (Fig. 5). A further increase in nitrate concentration did not have any significant effect on the branch growth.

In ‘Cleopatra’, adding 8 mM nitrate to the water with the highest chloride content (48 mM) caused a 51% reduction (from 13.8 to 6.7 nmol·g⁻¹ fresh weight) in the elevated level of PUT induced by the chloride (Table 1). The effect of nitrate addition was more pronounced in ‘Troyer’, where a 73% reduction (from 232.5 to 63.8 nmol·g⁻¹ fresh weight) in PUT level was recorded upon increasing the nitrate concentration from 2 to 8 mM in irrigation water containing 48 mM chloride (Table 1). Nitrate supplementation did not affect SPD levels in the leaves of either rootstock (Table 1), nor did it affect the SPM level in ‘Cleopatra’ leaves. However, a decrease of about 60% in the SPM level in ‘Troyer’ leaves was recorded at the low (2 mM) and medium (16 mM) chloride concentrations upon increasing the nitrate concentration in the water from 2 to 8 mM (Table 1).

Fig. 3. Effect of chloride concentration in the irrigation water on putrescine, spermidine, and spermine content in the leaves of ‘Cleopatra’ and ‘Troyer’ plants. The irrigation water contained 2 mM nitrate. Note the different scales of the y axis among the figures. Data are mean of six replicates. Means within each rootstock with different letter are statistically different (P ≤ 0.05).
uptake and transport to the leaves and the ability of its leaf tissue ‘Cleopatra’ may be attributed to its ability to restrict chloride showed severe damage (Fig. 1). Thus, chloride resistance in damage was recorded in ‘Cleopatra’, whereas ‘Troyer’ leaves even at the lowest chloride concentration of 2 mM (Fig. 1), which is considered good-quality irrigation water. In contrast, the salt-sensitive rootstock ‘Troyer’ showed damage via K or Mg deficiencies can be ruled out. Adding chloride to the water. Therefore, effect of chloride on PUT levels did not appear to affect the toxic effects caused by PUT. The increased level of the diamine PUT and the reduced levels of the triamine SPD and the tetraamine SPM could be a result of the lower availability of the precursor propylamine moiety derived from S-adenosylmethionine (SAM). Competition between ethylene and polyamine pathways for the common precursor SAM has been shown (Even-Chen et al., 1982). Blockage of the SPD and SPM biosynthesis pathway could explain, at least in part, the PUT accumulation, assuming that PUT synthesis is maintained at a constant rate. SPD and SPM synthesis blockage could be attributed to a reduction in the availability of the propylamine moiety due to increased ethylene production. Recent studies (data not shown) revealed increased ethylene production in a salt-sensitive citrus rootstock in response to high chloride levels and a small decrease in ethylene evolution of a salt-tolerant rootstock. Whether PUT is a causal agent in the expression of stress symptoms or a consequence of the stress has not yet been established. Strogonov et al. (1972) suggested the toxic effects observed in salt-stressed plants are caused by the elevated salt content and the accumulation of PUT in the tissue. The results from the present study with applied polyamines support Strogonov’s suggestion but not the suggested protective role of SPM (Altman, 1989; Young and Galston, 1983). Potassium and magnesium deficiencies are known to increase the PUT titer (Corey and Barker, 1989; Smith, 1984). In our study, K and Mg levels in the leaves were not affected by the chloride content in the water. Therefore, effect of chloride on PUT levels via K or Mg deficiencies can be ruled out. Adding chloride to the irrigation water did not significantly affect N level in the tissues of either rootstock. Therefore, it can be concluded that chloride at high levels does not exert its deleterious effects via the alteration of N content in the tissue.

Addition of nitrate to irrigation water with high concentrations of chloride resulted in reduced chloride accumulation in the leaves (Fig. 4) and, therefore, could account for the resultant alleviation of the toxic effects inflicted by high chloride levels (Fig. 5). The beneficial effect of increased nitrate nutrition was apparent in both rootstocks, but more pronounced in the chloride-sensitive ‘Troyer’ which displayed severe toxic symptoms of growth inhibition, leaf and branch-tip burns, and high PUT titer in response to increased chloride concentrations. The decrease in PUT level caused by enhanced nitrate nutrition could be considered as contributing to
the reduction of the toxic symptoms observed under high chloride conditions.

Although increasing nitrate concentration in the irrigation water caused a decrease in chloride level in the leaves, increasing chloride concentration in the water did not affect N level in the leaves. The nitrate anion taken up by the plant is metabolized and the N incorporated to organic compounds, transferring the negative charge to organic anions, while the chloride is not metabolized and maintains its negative charge (Cram, 1973). Thus, the chloride is accumulated within the plant as opposed to the nitrate anion. As a consequence, nitrate is preferentially absorbed by the plant over chloride. In addition, the presence of nitrate in the nutrient solution can inhibit the chloride flux into the root cells (Glass and Siddiqi, 1985) due to allosteric influence of nitrate on the site of chloride uptake (Cram, 1973).

The detrimental effects of chloride on citrus can be attributed to either an increase in the osmoticum of the nutrient solution or to the chloride ion’s specific properties. Our results with ‘Troyer’ grown in iso-osmotic solutions with reciprocal chloride : nitrate molar ratios (2:16 vs. 16:2 mM) showed toxic symptoms only when chloride was the main anion (Fig. 5). It is suggested that, in the range of chloride concentrations tested, the toxic symptoms are not due to osmotic effects, rather to the chloride ion’s toxic characteristics. The results obtained here suggest that adding nitrate can effectively alleviate chloride damage in citrus orchards irrigated with high-chloride containing water.

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Table 1. Effects of nitrate supplementation to the irrigation water on putrescine, spermidine and spermine levels in leaves of ‘Cleopatra’ and ‘Troyer’ plants irrigated with water containing increasing concentrations of chloride. Data are mean of six replicates. Means within each column with different letter are statistically different (\( P \leq 0.05 \)). All the differences between rootstocks are statistically different (\( P \leq 0.01 \)).

| Cl : NO\(_3\) (mM) | Putrescine | Spermidine | Spermine |
|---------------------|------------|------------|----------|
|                     | Troyer     | Cleopatra  | Troyer   | Cleopatra  | Troyer   | Cleopatra  |
| 2 : 2               | 48.4 bc    | 5.1 b      | 512 a    | 1428       | 55.5 a   | 1409       |
| 2 : 8               | 37.0 c     | 5.5 b      | 303 b    | 1455       | 22.5 c   | 991        |
| 2 : 16              | 19.3 cd    | 6.8 b      | 437 ab   | 1440       | 20.9 c   | 929        |
| 16 : 2              | 56.0 b     | 6.5 b      | 408 b    | 1543       | 50.7 a   | 1111       |
| 16 : 8              | 25.9 cd    | 6.0 b      | 362 b    | 1549       | 20.4 c   | 996        |
| 16 : 16             | 16.1 d     | 6.1 b      | 415 b    | 1645       | 24.8 bc  | 840        |
| 48 : 2              | 232.5 a    | 13.8 a     | 406 b    | 1499       | 31.1 bc  | 1110       |
| 48 : 8              | 63.8 b     | 6.7 b      | 373 b    | 1892       | 41.6 b   | 1012       |
| 48 : 16             | 33.8 c     | 5.2 b      | 406 b    | 2375       | 40.0 b   | 998        |

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