Research Article

Investigation on Heat Transfer Enhancement in Microchannel Using Al₂O₃/Water Nanofluids

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Nowadays, reducing heat generation in electronic devices while using microchannel cooling is used to solve this problem. Because the trend is globally marching toward the compact size, the component’s dimensions get smaller, but the warmth involved within the component increases. Studies of heat transfer rate are conducted to determine the effect of a fully heated microchannel conductor’s heat transfer performance. Experiments are performed using nanofluid Al₂O₃/water through a concentration percentage of 0.1% and 0.25% and deionized water through a microchannel conductor with 25 rectangular microchannel numbers with a dimension of (0.42 × 0.42 × 100) mm³. This present work deals with the effect of nanofluids and their concentration percentages. Finally, it concluded that better heat transfer performance was seen in nanofluids compared to deionized water. The reason is the high viscosity of nanofluid Al₂O₃/water due to these nanoparticles is deposited on the wall surface of the microchannel and outcomes trendy improvement in the heat transfer. Finally, a high concentration percentage of nanofluids revealed a practical improvement in the transfer of microchannel. As a result, 0.25% of the concentration percentage achieved a satisfactory result compared to the remaining fluids and almost 32.5% and 26% of thermal resistance decrease.

1. Introduction

The microchannel is one of the significant materials to reduce the channel’s size to the small resolve reason rise in heat transfer rate. The microchannel is an essential material used to reduce the channel’s size to a small resolution to increase heat transfer [1]. Made of silicon-based microchannels, the Pyrex cover plates are bonded to the anode and maintain constant heat flux. The maximum of 2100 is the Reynolds number, the heat transfer rate of 760 W/cm², and the 2 bar is the pressure drop [2]. It was found that the microchannel conductor will be prepared to dissipate over
the water inlet temperature with a maximum substratum temperature increase of 7°C. Flow maldistribution depends on several factors like device geometry (cross-section and inlet and outlet shapes, sizes, engineering tolerances, or inadequacies) and functioning circumstances (some of the variation of fluid flow rate, viscosity, and thermophysical changes) [3]. A flow guide plate has been added to scale back the flow maldistribution and extend the heat mass transfer. It had been found that flow maldistribution and average temperature decreased while increasing the number of fluid guides [4]. Extreme maldistribution is observed at the header with circular header configuration, while flow through all channels was almost homogenous with a tapered header configuration [5]. When comparing circular cross-section and tapered cross-section, the sharp cross-section header displayed less fluctuation than the circular cross-section header indicating additional constant channel flow [6]. Within a simulation study, eight strategies were considered to know uniform flow distribution, and it had been identified. The only design to know was the direct enlargement of the distribution header [7]. Higher Reynolds numbers were advantageous but with a more significant pressure drop for the heat sink. While observing the temperature distribution within the heat sink, the numerical predictions and experimental data matched well [8, 9]. Numerically and experimentally, water-cooled microchannel heat sink performance and found a new correlation incorporating the effects of both channel size and coolant mass velocity to improve pressure drop prediction [10]. The temperature-dependent thermophysical property method was used to investigate the effect of microchannel measurements and, consequently, properties like fluid flow and physical thermo [11]. The coefficient of heat transfer and the microchannel conductor’s thermal resistance was compared with the experimental results and located to have good agreement [12].

In metallic copper channels, two types are single and multichannel and have better heat transfer performance and pressure drop by using R134a refrigerant and water. Theoretical results of R134a refrigerant and water were compared to experimental results [12]. Heat transfer characteristics in rectangular microchannels to explore the validity of theoretical correlations. The experiment was performed with water and FC770. The following factors are essential for enhancing nanofluid performance: size, material, volume, temperature, and dispersant [4]. The heat transfer performance is affected by the base fluid, material of nanoparticles, volumes, temperature, size of nanoparticles, and flow properties [13]. The total loss in pressure was calculated for the flow rate and rate of flow and ratio of sectional area to the valve and n number of channels. In the heat exchanger, the flow rate was enhanced based on total pressure loss [14]. The enhancement in heat transfer while using copper/water nanofluid was carried out with the help of a fin, which has 100 and 57-nanometre dimensions. It was observed that the heat transfer coefficient was significantly increased due to existing nanoparticles. Heat transfer enhancement is gradually increasing when thermal conductivity and thermal resistance are increasing. One crucial factor is that there is no extra pressure drop when using nanofluids because of their size and volume [15, 16]. Nanofluids through metal nanoparticles exhibited better cooling performance than nanofluids with nonmetallic particles owing to the developed thermal conductivity of metal nanofluids. When the volume concentration and Reynolds number increased, the nanofluids would increase the heat transfer [17]. It had also been observed that there was no noticeable increase within the friction factor by adding nanoparticles into the clean water. The best heat transfer improvement remained aimed at CuO/water nanofluid for given values [18]. The type of base fluids mainly influenced the heat transfer augmentation of nanofluids, the ratio of ethylene glycol: water mixture, the material of nanoparticles, volume concentration, temperature, nanoparticle size, and flow characteristics [19]. The friction, pumping power got considerably increased with the utilization of nanofluid. The ratio of Brownian movement and thermophysical effects had relatively significant effects [20]. The rationale was that the warmth transfer between fluid and channel walls intensified due to interruption of flow by the thermal gesture of nanoparticles. Hydrodynamic and warmth transfer behavior of nanofluid was predicted well while considering the slip length with nanofluid thermophysical properties [21, 22]. The friction factor decreased with the Reynolds number, thermal resistance decreased with pumping power, and input heat in a rectangular microchannel heat sink made of copper using water as a working fluid [23]. A conventional analysis approach can predict heat transfer behavior in microchannels since the numerical predictions showed good agreement with the experimental data. A single rectangular microchannel heat sink is with Al2O3 as nanoparticle and base fluid such as water and ethylene glycol with various volume concentrations 0.6%, 1.2%, and 1.8% [24]. Experiments were carried out to investigate the effect of volume concentrations of the nanoparticles on convective heat transfer and fluid flow in microchannels. Appreciable enhancement of the convective heat transfer coefficient was found with nanofluids, with the base fluid of water and ethylene glycol without significant friction loss [25].

A single rectangular microchannel heat sink is with CuO/water nanofluid with various volume concentrations such as 4.5%, 1.03%, and 0.24% under both laminar and turbulent conditions. For nanofluid, there was a slight heat transfer enhancement with low particle volume concentration. For a given Reynolds number, nanofluid overall energetic performance, defined by heat transferred/pumping power ratio, remained lower than water [26]. The decrease in the ratio was due to the augmentation of the particle volume concentration. The heat transfer performance and pressure drop of nanofluids in a microchannel heat sink are with TiO2/water nanofluid of different volumetric concentration ratios of 0.25%, 0.5%, 1.0%, 1.5%, and 2.0% to analyze the upper limitation of the particle volume concentration for heat transfer performance [27]. It was identified that by using nanofluid up to 2.0%, the heat transfer increased, and the heat transfer decreased after the volumetric ratio of 2.0%. The thermal resistance of nanofluid decreased by adding nanoparticles with an average diameter smaller than 25 nm into the base fluid [28]. In a single microchannel with
two types of nanofluid such as Al2O3/water and SiO2/water with a volumetric concentration of 0.005%, 0.01%, and 0.02% and found Al2O3/water had a significantly high heat transfer coefficient. It was seen that Al2O3/water nanofluid had more extraordinary ability in disrupting the thermal boundary layer profiles and lead to a higher percentage reduction in thickness of boundary layers. Al2O3/water showed a higher heat transfer coefficient than the SiO2/water and the base fluid [23, 29, 30]. The various performance values of microchannels with various nanofluids are listed in Table 1. According to Table 1, there are so many moderate values in heat transfer rate, thermal resistance, and pressure drop based on nanofluid type.

According to the literature studies, it was found that there is limited researchers carried out with the combined effect of nanofluid with microchannel; also, no results were found for various inputs. Hence, the heat transfer rate in the microchannel, the consequence of Al2O3 with water nanofluids is investigated with various power outputs and because of optimum cost this Al2O3/water will frequently act as a coolant for small photovoltaic applications. The scanning electron microscopy is carried out to analyze the performance enhancement of the microchannel conductor performance under thoroughly heated conditions has been experimentally investigated.

2. Preparation of Nanofluid

Nanofluids are prepared using Al2O3 (alumina) particle blended with water through a magnetic stirrer at a rate of 1000 rpm, as shown in Figure 1. Mixed Al2O3 particles and distilled water were also processed at 60°C in the ultrasonic bath for 60 minutes, as shown in Figure 2 with 490 kW and 18 kHz frequency. Two concentration percentages of 0.1% and 0.25% have been used in this nanofluid preparation. The JSM IT-800 SEM (scanning electronics microscope operating at 5 kV) image is pure nanoparticles, as shown in Figure 3. The required alumina was purchased from nano-shell, India, which is used to prepare a nanofluid. It was found that 40 nm is the average particle size by using Sherrer’s equation.

\[
d = \frac{0.9\lambda}{\beta \cos \theta},
\]

Pak and Cho’s equation (2) is used to calculate the nanofluids density,

\[
\rho_{nf} = \rho_s (1 - \varphi) + \varphi \rho.
\]

The Maxwell equation (3) is used to find out the thermal conductivity,

\[
\frac{K_{nf}}{K} = \frac{K_s + 2K + 2\varphi (K_s - K)}{K_s + 2K - \varphi (K_s - K)}.
\]

Nanofluid’s specific heat is calculated using Xuan and Roetzel’s equation

\[
(\rho c_p)_{nf} = (1 - \varphi)(\rho c_p) + \varphi(\rho c_p)s.
\]

Viscosity plays a main role in fluids. This viscosity is calculated by using Einstein equation

\[
\mu_{nf} = \mu (1 + 2.5 \varphi).
\]

The pH upsides of the prearranged nanofluids are estimated by using digital pH -14L and discovered to associate with 5, which is a long way from the isoelectric point of 9 for alumina nanoparticles. This guarantees that the nanoparticles are all around scattered, and the nanofluid is steady a direct result of huge loathsome powers among the nanoparticles when pH is a long way from isoelectric point. At isoelectric point, the loathsome powers between particles are excessively feeble, permitting the particles to move toward one another and in the long run agglomerate, influencing the solidness of the suspensions. Moreover, Zeta potential has been estimated using Japanese nanomolecule analyser and discovered to be +43 mV to +47 mV, which is demonstrative of acceptable strength of the nanofluid. A pH worth of 3.7 to 5.2 will keep the nanofluids stable for a long time.

The properties of working fluid are found as illustrated in Table 2.

3. Experimental Setup

The microchannel cooling system utilized within the experimental study is displayed in Figure 4. Set up comprises test section, pump, liquid reservoir, pressure transducer, computerized system to store the data, and a collecting tank. The channel size is identical to that of flow distribution studies. An oblong header with a width of 1 cm, depth of 0.5 cm, and length of 19.6 cm is used, as shown in Figure 5. The test section of the microchannels is shown in the schematic view.

To look at the fluid flow within the microchannels, the examination device is fixed in the roofed clear board. The lowest layer that is the examination portion was formed to improve the heater. Therefore, the test section has been well protected from ameliorating the heat energy loss. The microchannel conductor’s surface heat flow and the inlet and outlet heat flow for a working fluid are calculated using temperature sensors. The thermocouples are fixed at the test section at various places and a total of 10 thermocouples. Eight numbers are in aluminum blocks with an equal distance of 50 mm at the edges of both sides in the bottom layer of the microchannels conductor to reside the surface temperatures monitor the heat flow. Experiments are performed in a microchannel using three types of fluids: two Al2O3/water nanofluids with concentration percentages of 0.1%, 0.25%, and deionized water are conducted to measure the microchannel conductor’s pressure drop and heat transfer performance.
4. Experimental Procedure

The test was initiated by a heater connected to the bottom section of the microchannel. The test section consists of the outlet header and the collection container used for fluid flow and storage purposes. Until the beginning of the experiment, the temperature sensors are set to maintain the water bath at a constant temperature. Pressure transducers can calculate the fluid flow state of the inlet and outlet pressure and the difference. In this work, fluid and surface temperatures were monitored every 5 seconds through a computerized data acquisition system. Initially, microchannels inspect the leakage of the test segment by applying water

| Type of nanofluid used | Heat transfer rate (W/m²K) | Thermal resistance (K/W) | Pressure drop (N/m²) | Reference |
|------------------------|-----------------------------|--------------------------|----------------------|-----------|
| SiO2–based dilute nanofluids | 153 | — | 0.0020 | [5] |
| Ethylene glycol: water nanofluids | 165 | 1.1 | 0.0022 | [23] |
| Cu-water nanofluid | 144 | 1.0 | 0.0024 | [3] |
| Water-Fe3O4/CNT hybrid nanofluid | 160 | — | 0.0025 | [18] |
| Oil/MWCNT nanofluid | 140 | 0.85 | — | [20] |
| Water/CuO nanofluid and L-shaped porous ribs | 150 | 0.7 | 0.0028 | [16] |
| Al2O3-water nanofluids | 165 | 0.6 | 0.0030 | [14] |
| Hybrid nanofluid | 157 | — | 0.0031 | [2] |
| Graphene nanoplatelets–silver/water nanofluids | 168 | 0.83 | 0.0023 | [27] |
| FMWCNT nanofluids | 146 | 0.81 | — | [28] |

Figure 1: Preparation of nanofluid using a magnetic stirrer.

Figure 2: Preparation of nanofluid using an ultrasonic bath.

Figure 3: SEM image of Al₂O₃ nanoparticle.
immersion and pressurized air. After that, the fluid movement with or without a channel ensures the passage of potash permanganate—a water solution. Testing was performed for various heat input conditions and fluid flow rates. The system has come to a steady-state. It has taken up to 1 hour. In this research work, all data on the measured pressure and temperature of the microchannel were stable, and the fluid temperature of 30°C was maintained throughout the entire experiment. To ensure the repeatability test, the investigational analysis was replicated, aimed at several occasions taking place various beings.

5. Uncertainty in the Experiments

The experiments are conducted with the following uncertainty limits as shown in the table, calibrated using the visual inspection machine system in Table 3. It was found that ±1.71% of uncertainty was obtained for the microchannel with the dimension of 0.42 mm width and depth, 100 mm length, and 10 mm of width, 5 mm of depth, and 196 mm of length. These uncertainty values depend on the errors, which all are in the measured quantities. Based on the measured quantity values, it was observed that ±0.2% is the error value for thermal resistance and ±3.5% for Reynolds number. The same error and uncertainty values are calculated for both inlet and outlet temperatures and base temperature also, and it was observed that ±0.2% and ±2.36%.

6. Result and Discussion

In this research work, the microchannel conductor is a device used to determine the performance of a heat transfer rate using nanofluid Al₂O₃/water at a concentration of 0.25%, 0.1%, and deionized water with various heat inputs such as 6 W, 12 W, and 15 W.

Analysis carried out plotting the graphs with a heat transfer coefficient, a dimensionless coefficient, thermal resistance with pressure drop are contrasted with variable Reynolds number to compare the results of deionized water with Al₂O₃/water nanofluids.

6.1. Influence Impact of Heat Transfer Rate. Present work and the heat transfer rate in the microchannel conductor by applying nanofluids were shown in Figures 6–8. Al₂O₃/water nanofluids are 0.1%, 0.25%, compared to the deionized water heat transfer coefficient versus the different conditions of the Reynolds number. From the results obtained, the maximum and minimum heat transfer coefficients are in Al₂O₃/water nanofluids (0.25%) and deionized water for all conditions of Reynolds numbers. Improved nanofluids are discussed with water due to significant thermal conductivity and heat capacity, and thermal dispersion in nanoparticles. As a result, high thermal power in nanofluid contrasts with deionized water increases heat transfer rate. Contrasted to deionized water, significantly improved heat transfer is 37% while also nanofluid Al₂O₃/water at a concentration percentage of 0.25% by volume at the considered absolute best Reynolds number.

6.2. Influence of Impact in Concentration Percentages in Nanofluids. The impact of the concentration of nanofluid percentages (0.1%, 0.25%) on the microchannel conductor's
performance was shown by plotting the dimensionless number relative to the Reynolds number as exposed in Figure 9. The mean heat transfer coefficient observed in the Al$_2$O$_3$/water nano fluid improved by 0.25% as per the plotted graph. Al$_2$O$_3$/water nano fluid of percentage concentration of 0.25% and 0.1% revealed the 32.8% and 23.16% improvement in heat transfer performance. Nanoparticle aggregation and accumulation are observed at high percentage concentrations of nano fluids, and therefore, the amount concentration within this study is limited to 0.25%.

6.3. Influence of Impact in Thermal Resistance in Nano fluids. Figure 10 shows the microchannel conductor thermal resistance concerning different Reynolds number conditions using Al$_2$O$_3$/water nanofluids (0.1%, 0.25%) and deionized water. Using obtained outcomes, maximum and minimum thermal resistance is deionized water and Al$_2$O$_3$/water of 0.25%, respectively. Overall, nanofluids have less thermal resistance than deionized water, which is often representative of lower microchannel wall temperature vital to the heat dissipation of photovoltaic module concentration and electronics devices. Almost 32.5% and 26% of thermal resistance decrease in Al$_2$O$_3$/water by 0.25% and 0.1% compared to deionized water [31–36].

6.4. Influence of Impact in Pressure Drop in Nano fluids. Figure 11 shows the microchannel conductor pressure drop
concerning different Reynolds number conditions using Al₂O₃/water nanofluids (0.1%, 0.25%) and deionized water. When rising the Reynolds number, increases of the pressure drop of fluids were noticed. Overall, minimum pressure drop was observed in deionized water compared to nanofluids. Because nanofluids have high viscosity and the particles of nanofluids are deposited on the channel’s surface, it leads to the roughness of the channel surface. Nanofluids have enhanced thermal management applications in electronic devices and many other purposes due to their significant high-pressure drop than deionized water. To achieve this, they require the overall performance of the heat transfer. Even so, there is increasing pressure drop in the microchannel at the time of practical applications because of deposition of solid particles [36–40].

7. Conclusion

In this research, the experiment carried out in the microchannel conductor using Al₂O₃/water nanofluids at a percentage concentration of 0.25%, 0.1%, and deionized water resulted in a recital of heat transfer rate. The findings from the results are

(i) Al₂O₃/water nanofluid of percentage concentration of 0.25% and 0.1% revealed the 32.8% and 23.16% improvement in heat transfer performance heat transfer augmentation by Al₂O₃/water is higher than deionized water owing to developed effective thermal conductivity. Hence, Al₂O₃/water is a suitable coolant for the thermal management of photovoltaic applications

(ii) Almost 32.5% and 26% of thermal resistance decrease in Al₂O₃/water by 0.25% and 0.1% compared to deionized water

(iii) Nanofluid concentration percentages are proven to obligate a more significant influence on the microchannel’s heat transfer performance. Nanofluid concentration percentages play a vital role in the microchannel recital of heat transfer. High concentration percentages of nanofluids achieved satisfying results due to more substantial thermal dispersion and thermophysical effects

(iv) It was found that this has been considered at the time of microchannel designing for sensitive temperature applications like cooling of photovoltaic cells, cooling of electronic chips, etc.

(v) Consequent nanofluid-induced pressure drop is needed not high that nanofluids are suitable chilling liquids intended for microchannels

Data Availability

The data used to support the findings of this study are included in the article.

Conflicts of Interest

The authors declare that there is no conflict of interest regarding the publication of this article.

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