Hydrochar as A Vehicle for Phosphorus Cycling from Dairy Manure to Cropland

B Brian He* and Lide Chen

1Chemical & Biological Engineering, College of Engineering, University of Idaho, 875 Perimeter Drive MS 2060, Moscow, ID 83844, USA
2Soil and Water Systems, Twin Falls Research and Extension Center, University of Idaho, 315 Fall Avenue East, Twin Falls, ID 83303, USA

Abstract

Dairy manure contains high concentrations of unutilized phosphorus which is not only a waste of resources but also a burden to dairy farmers and a threat to our environment. Phosphorus cycling and reuse would have a great impact on the dairy industry and agricultural economy as whole. Hydrochar possesses better biological properties when used for soil improvement and carbon sequestration. When produced from dairy manure, hydrochar can attain more than 90% of the total phosphorus content in manure and thus serves as an efficient vehicle for recycling nutrients from waste streams back to the crop system. Research findings prove that hydrochar from dairy manure is scientifically feasible and technologically promising as a vehicle for phosphorus recycling from waste to cropland.

Introduction

World consumption of P₂O₅ for fertilizer and industrial uses was 47 million tons in 2020 and the reserve of phosphorus rock was more than 300 billion tons [1]. Phosphorus rock is mainly found in sedimentary phosphate deposits and some as igneous phosphate deposits [2]. Although phosphorus cycles biogeochemically through the lithosphere, hydrosphere, and biosphere, its original source is non-renewable. Therefore, it is a logically wise strategy to utilize limited phosphorus reserves effectively and efficiently for generations to come.

The dairy industry provides quality food for our societies but also produces tremendous quantities of solid dairy manure. In the U.S., there were 8.86 million heads of milking cows in 2020 [3], which translates to over 240 million metric tons (530 billion pounds) of dairy manure generated annually, estimated based on lactating cows [3]. Phosphorus utilization by dairy cows is not in high efficiency [4,5]. The phosphorus produced in the dairy manure of a lactating cow is in the range of 77-95 g (0.17–0.21 lb) per day [4], which is equivalent to approximately 28-35 kg (62-76 lb) of phosphorus per year per cow or 248,000-310,000 metric tons (549-679 million pounds) of phosphorus by the 8.86 million cows in the U.S. in 2020. This is not only a waste of our limited resources but also creates a tremendous burden for dairy farmers and for our environment. Therefore, it would have a beneficial impact on agriculture, both dairy and crop systems, if the phosphorus can be even partially recycled from dairy manure and reutilized back to cropland.

Dairy manure contains nutrients that are needed by crop systems. However, nutrient reutilization of dairy manure through direct land applications as a fertilizer has long been deemed inefficient and imbalanced [6,7]. Soil near confined dairy operations can only uptake limited quantities of directly applied manure due to the major concern of nutrient leaching to surface and ground waters. Dairy manure, even after treatment by traditional lagoon systems, cannot be transported over long distances for nutrient distribution in an economically feasible way. Therefore, carefully designed ma-
Hydrochar via Hydrothermal Carbonization

Hydrochar is a carbonaceous material produced thermochemically from organic matter or biobased resources. The specific production process of hydrochar is commonly known as hydrothermal carbonization (HTC). This is largely due to the fact that HTC is typically conducted in aqueous solutions under moderate temperature (180-250 °C) and autogenous pressures (1.5-4.5 MPa) in an enclosed system [12,13].

HTC converts organic matter in an oxygen-absent or oxygen-lean environment in aqueous slurries through a series of complex reactions of hydrolysis, condensation, dehydration and decarboxylation, of which the mechanisms are not yet completely understood [14-16]. Processing of biomass via HTC yields a solid hydrochar as the targeted product, along with a post-processing liquid stream containing various water-soluble components. Since HTC processing does not require dry matter as the feedstock, it is distinctively advantageous to process wet biomass, such as dairy manure, directly without pre-drying [17-19].

There is a major difference between what is commonly known as biochar and hydrochar. Biochar is the name mainly used for the byproduct of a fast pyrolysis process of biomass or the product purposely produced from biomass for various applications, requiring dry biomass as the feedstock [20-23]. Evidently, biochar has totally different targeted applications, such as adsorbents for wastewater treatment and soil amendments and fertilizer [24,25].

Table 1: Carbon and nutrient attainment rates at different temperatures with 15% of solid content & 60 min [12].

| Operating Temp. (°C) | R\textsubscript{Carbon} | R\textsubscript{Phosphorus} |
|----------------------|------------------------|---------------------------|
| 240                  | 61.4                   | 83.5                      |
| 220                  | 66.3                   | 82.5                      |
| 200                  | 69.8                   | 92                        |
| 180                  | 74.4                   | 94.4                      |

It has been evidenced that hydrochar production can be achieved at a temperature as low as 180°C with 15-30 min of reaction time, and the reaction time can be shortened greatly by raising the temperature to 250°C. The retention rate of nitrogen in hydrochar is more sensitive to processing conditions than those of carbon and phosphorus, decreasing as the operating temperature increases and reaction time extends. This phenomenon might be due to the decomposition of proteins into amino acids which in turn transfer to liquid phase [12,29].

It was found that under higher temperatures (e.g., 250°C), the yield of hydrochar is 78.8% at 15% of solid content and 68% at 5% of solid content. In contrast, the hydrochar yield is approximately 86% at 5% and 2% of solid contents if the operating temperature is at 200°C or 180°C for 60 min [12].

It was concluded that hydrochar yield from dairy manure is governed by the combination of process parameters, including processing temperature, retention time and solid content of the feed manure. Further investigation of process optimization revealed that an optimal phosphorus attainment of 90% or higher in hydrochar can be achieved under a set of processing condition of 224 °C of processing temperature, 30 min of processing time and 9.6% of solid loading rate [32]. Although varying under different processing...
conditions, the phosphorus attainment rates in hydrochar can be as high close to 100% [32]. Thus, hydrochar has a greater potential of retaining the nutrient values carried over from dairy manures. This agrees with the conclusions by others [31,33].

Phosphorus readily available to plants must be in water-soluble phosphates. However, among the total phosphorus in manure, inorganic phosphorus is soluble and mainly in the liquid fraction, and organic phosphorus in the solid fraction [5,34]. Approximately 73% of phosphorus in fresh livestock wastes are in organic form [35], such as phospholipids and phytic acids (inositol hexakisphosphate, \( \text{C}_6\text{H}_{11}\text{O}_{14}\text{P}_7 \)), which requires microbial activities, such as phosphatases and phytase, to generate free phosphates that plants can utilize.

Hydrochar retains high levels of total phosphorus in manure. Although more scientific studies are needed before one can have a clear picture of the fate of phosphorus in hydrochar, it has been found that the hydrochar production process via HTC alters chemically the organic phosphorus in manures into ortho-phosphates [33,36].

Organic phosphates in the liquid are deposited into the solid hydrochar and all calcium phosphate phase converted to crystalline hydroxylapatite at 225 °C. It was believed that thermochemical processing is responsible for the transformation mechanisms in HTC. The affinity cation ions to phosphorus largely determine the phosphorus speciation evolution during HTC treatment [36]. It was also found that cation ions such as Ca, Al, and Fe that form precipitating phosphates may be the most important factor which in turn determines the potential of phosphate retention in the solid phase. The phosphorus change through HTC is tremendously advantageous in converting the unavailable organic phosphorus for better crop utilization.

In summary, phosphorus in dairy manure can be recycled via hydrochar through the HTC process. Hydrochar production from dairy manure not only provides an alternative way of manure management but also creates a value-added product for dairy farmers. Hydrochar is an excellent vehicle that attains the phosphorus cycling becomes a reality.

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Conflict of Interest

No conflict of interest.

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