Determination of Punicalagin Isomers in Pomegranate Husk

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Abstract

Punicalagins are the main ingredients of phenolic compounds in pomegranate (Punica granatum L.) husk. A simple and accurate method for punicalagin analysis based on ethanol extraction and RP-LC using linear gradient of methanol in 0.1% TFA solution was established. The feasibility of this procedure was tested by analyzing the punicalagin level both in fresh pomegranate husk collected from different provinces in China and dried husk from a drugstore. The content of each isomer and total content of punicalagins in husk were determined. The mean value of punicalagins content in pomegranate husk was 82.4 mg g⁻¹. The highest content of punicalagins was found in a variety of husks from Shanxi province, while the lowest content was found in the husk from the drugstore in Guangdong province.

Keywords

Column liquid chromatography
Pomegranate husk
Punicalagin

Introduction

Pomegranate (Punica granatum L.) husk, a traditional Chinese medicine used as antibacterial, anti-inflammatory and hemostasis agent, is rich of phenolic compounds [1]. The most abundant of these polyphenols is punicalagin.

Punicalagin is a potent antioxidant. The potent bioactivity of punicalagins can be explained by its ability to hydrolyze into ellagic acid (EA) in vivo and across the mitochondrial membrane in vitro. It was also found to protect against oxidative damages to lipids, amino acids and guanosine as [2].

Ellagic acid is a dietary polyphenol found in fruits and nuts and implicated with potent antioxidant, anticancer and antiatherosclerotic biological properties. Free EA in human plasma is due to its release from the hydrolysis of ellagitannins such as punicalagins [3].

Due to the importance of this compound, methods such as liquid chromatography (LC) [4–6] and LC-MS [7] have been used to analyze punicalagin in pomegranate extracts. Punicalagins are often noted in the singular punicalagin, however, it is found naturally as two reversible α- and β-anomers. Considering this aspect, punicalagins are difficult to analyze quantitatively.

Furthermore, all reported methods were described as time consuming and were incapable of screening a large number of samples without elaborate sample preparation. For example, they need purification by Sephadex-LH-20 [4] and sample preparation was by Toyo-
pearl HW 240C column [6]. Reported methods gave the content of punicalagin in extracted samples, but not the real content in pomegranate husk. A rapid, simple and accurate analytical method is reported here to determine the content of each punicalagin isomer and total content of punicalagins in pomegranate husk. Punicalagins in 16 varieties of pomegranate husk were measured.

**Experimental**

**Sampling**

Fourteen varieties of fresh pomegranate husk were collected from seven provinces in China and two varieties of dried pomegranate husk were from drugstores. Punicalagin isomer standards were obtained by HSCCC as we reported before [8]. Ethanol used for sample preparation was of analytical grade and purchased from Beijing Chemical Factory (Beijing, China). Trifluoroacetic acid (TFA) used for LC analysis was of chromatography grade and purchased from Merck (Hohenbrunn, Germany). Methanol used for LC analysis was of chromatography grade and purchased from Dima (Dima Technology Inc., USA).

**Sample preparation**

Fresh pomegranate husk was dried naturally at room temperature then grounded to powder. One gram of dried husk powder was extracted ultrasonically with 30 mL 40% ethanol for 30 min twice. The supernatants were combined, ethanol was evaporated under vacuum. The extract was freeze-dried, yielding about 0.5 g crude sample. Thirty milligram of the sample were precisely weighed by a Sartorius scale (BP211D) and dissolved in a 10 mL volumetric flask with methanol-water (1:1). Then it was filtered through a 0.22 μm membrane filter prior to injection.

**Operating Conditions**

The LC used was a Shimadzu LC-20AVP system with two LC-20AT solvent delivery units, an SPD-20A UV-vis detector, a CTO-10ASVP column oven (Shimadzu, Kyoto, Japan), a T2000P workstation (Beijing, China) and a reversed phase C18 column (250 × 4.6 mm, 5 μm, DiamodsiT™).

The solvent system consisted of MeOH (eluent A) and 0.1% (v/v) TFA in water (eluent B). Gradient conditions: 0–10 min, 5–20% A in B; 10–25 min, 20–45% A in B; 25–35 min, 45–65% A in B; 35–38 min 65% A in B. This was followed by a 10 min re-equilibration. The column oven temperature was set at 30 °C. The flow rate was 1.0 mL min⁻¹, and 5 μL portions were injected into the column. Punicalagin was detected by absorbance at 378 nm. The representative chromatograms of the LC elution profile of punicalagin standard and that from pomegranate husk sample is shown in Fig. 1.

**Preparation of the Standard Curve**

Quantification was based on the external standard method. A stock solution was prepared with 10.0 mg of punicalagin reference standard, which was dissolved and diluted to 5 mL with methanol–water (1:1). 2, 3, 4, 5, 6, 8 μL were injected in duplicate. Peak areas were recorded for all the solutions.

**Statistical Analysis**

All data of total punicalagin content are presented as mean ± standard deviations (SD). The equilibrium constant $K$ of β-punicalagin and α-punicalagin content was also calculated. It was defined as β-punicalagin content divided by α-punicalagin content. Standards and samples of each variety of pomegranate husks were dissolved in methanol, water and methanol–water 1:1 solution, respectively.
Results and Discussion

Linearity of Standard

Linear regression analysis of the peak area responses (y) versus the injection volume (x) gave the following equations: 
\[ y = 137746x + 141648 \] for z-punicalagin, 
\[ y = 224830x + 167999 \] for \( \beta \)-punicalagin and 
\[ y = 362576x + 309648 \] for total punicalagins. The correlation coefficients (r²) were 0.9983, 0.9995 and 0.9998, respectively, which demonstrated linearity of the method over the concentration range analyzed. Three correlation coefficients (r²) had significant differences after verification \( r_{0.05}(4) = 0.811; r_{0.01}(4) = 0.917 \). [9]

Precision and Recoveries

The system precision was determined by chromatographing six injections of the standard solution and calculating the relative standard deviation (RSD) of the peak area responses. The method precision was established by assaying six different extracts of the same husk sample with the proposed chromatographic method. The RSD for standard and samples were 1.05 and 0.74%, respectively. In our study, recovery is related to the system precision, so the RSD for standard and samples were 1.05 and 0.74%, respectively. We calculated the relative standard deviation (RSD) of the peak area responses. The method precision was determined by assessing the peak areas of standard solutions and samples.

Table 1 shows the total content (triplicate analyses) of punicalagins of 16 different varieties of pomegranate husk. The highest content of punicalagins in husk was found in the husk of a pomegranate named red ruby from Shanxi province (121.5 mg g⁻¹), which is nearly threefold the lowest content in husk from a drugstore in Guangdong province (39.8 mg g⁻¹). Data showed significant differences in the content of punicalagins among husks from various areas.

Analysis of Punicalagin Content in Pomegranate Husk

The LC method was used to quantitate punicalagin in a sample of dried pomegranate husk. Since pomegranate-derived foods are widely consumed and there have been conflicting reports regarding the toxicity of punicalagin [10, 11], there is a need for in depth in vitro and in vivo studies to determine the biological properties of this compound. Thus, it is important to analyze the content of punicalagin in the traditional Chinese medicine-pomegranate husk. Table 1 shows the total content (triplicate analyses) of punicalagins of 16 different varieties of pomegranate husk. The highest content of punicalagins in husk was found in the husk of a pomegranate named red ruby from Shanxi province (121.5 mg g⁻¹), which is nearly threefold the lowest content in husk from a drugstore in Guangdong province (39.8 mg g⁻¹). Data showed significant differences in the content of punicalagins among husks from various areas.

Equilibrium Constant of Punicalagin Content in Pomegranate Husk

It is reported that z-punicalagin and \( \beta \)-punicalagin interconverted and the equilibrium constant, \( K = [\beta]/[z] \) was one in methanol and four in water [11]. But in our study, the ratio of \( \beta \)-punicalagin and \( z \)-punicalagin content was about 1.60. It is interesting to note that this ratio will not change with the change of solvent.

Conclusion

Punicalagin, which mainly exists in pomegranate fruits, such as pomegranate husk, pith and carpellar membrane [12], is of great interest because of its antioxidant activities. However, in our studies the punicalagin levels have shown to be different among various accessions of pomegranate husk. It is very important to select several varieties with high punicalagin levels from numerous pomegranate husk accessions in drugstores or supermarkets for the study. The described method was found to be rapid, linear, reproducible, specific, accurate and capable of analyzing a large number of samples without elaborate sample preparation.

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