**Direct Measurement of Mercury Deposition at Rural and Suburban Sites in Washington State, USA**

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**Abstract:** Because of mercury’s (Hg) capacity for long-range transport in the atmosphere, and its tendency to bioaccumulate in aquatic biota, there is a critical need to measure spatial and temporal patterns of Hg atmospheric deposition. Dry deposition of Hg is commonly calculated as the product of a measured atmospheric concentration and an assumed deposition velocity. An alternative is to directly assess Hg deposition via accumulation on surrogate surfaces. Using a direct measurement approach, this study quantified Hg deposition at a rural site (Pullman) and suburban site (Puyallup) in Washington State using simple, low-cost equipment. Dry deposition was measured using an aerodynamic “wet sampler” consisting of a Teflon plate, 35 cm in diameter, holding a thin layer (25 mm) of recirculating acidic aqueous receiving solution. In addition, wet Hg deposition was measured using a borosilicate glass funnel with a 20-cm-diameter opening and a 1 L Teflon sampling bottle. Hg deposition was estimated based on changes in total Hg in the aqueous phase of the samplers. Dry Hg deposition was $2.4 \pm 1.4$ ng/m²·h (average plus/minus standard deviation; $n = 4$) in Pullman and $1.3 \pm 0.3$ ng/m²·h ($n = 6$) in Puyallup. Wet Hg deposition was $7.0 \pm 4.8$ ng/m²·h ($n = 4$) in Pullman and $1.1 \pm 0.2$ ng/m²·h ($n = 3$) in Puyallup. Relatively high rates of Hg deposition in Pullman were attributed to regional agricultural activities that enhance mercury re-emission and deposition including agricultural harvesting and field burning. Hg concentration in precipitation negatively correlated with precipitation depth, indicating that Hg was scavenged from the atmosphere during the beginning of storm events. Because of their relative simplicity and robustness, direct measurement approaches such as those described in this study are useful in assessing Hg deposition, and for comparing results to less direct estimates and model estimates of Hg deposition.

**Keywords:** dry deposition; wet deposition; wet sampler; agricultural field burning; AIRPACT

**1. Introduction**

Mercury (Hg) has a unique ability for long-range transport in the atmosphere [1]. Combined with the tendency of Hg to bioaccumulate in aquatic biota, atmospheric deposition of Hg is a worldwide health concern [2]. Natural sources of Hg are geogenic and include releases from outgassing mantel and crustal materials, volcanoes, and geothermal regions [3]. Key anthropogenic sources of atmospheric Hg include coal combustion, artisanal and small-scale gold mining, metal production, and cement production [4]. Direct anthropogenic sources account for ~30% of total Hg emissions to the atmosphere, natural emissions account for ~10%, and re-emission of previously deposited Hg from oceans and soils make up the remaining ~60% [4,5]. A substantial fraction of deposited Hg ends up in aquatic ecosystems where it can accumulate in aquatic biota as toxic methylmercury [5,6]. As a result of widespread Hg deposition in North America, 6.6 million lake
hectares in the U.S. have fish consumption advisories in place due to elevated concentrations of Hg in fish tissue, and 28 states including Washington have statewide fish consumption advisories due mainly to Hg contamination [7]. An important source of Hg deposition in the Pacific Northwest region of the U.S., the geographical focus of this study, is the long-range transport of Hg emissions from Asia [8].

The atmospheric cycling of Hg is complex, but many key processes are known [1–3,9,10]. Atmospheric Hg is dominated by gaseous elemental Hg (GEM), which has a residence time of months, allowing for long-range transport in the atmosphere. GEM can be transformed to gaseous oxidized Hg (GOM) in the atmosphere in the presence of ozone, chlorine gas, hydroxyl radical and bromine compounds. Because of its greater solubility and reactivity, GOM has an atmospheric residence time of days to weeks. GOM, with its relatively low vapor pressure, can partition onto particulate matter to form particulate-bound Hg (PBM), which, like GOM, has a relatively short residence time that is dependent on particle size. The surface flux of Hg is bidirectional since deposited Hg can efflux back into the atmosphere as GEM. Hg deposition can occur as dry or wet deposition. Depending on season and setting, dry deposition can be dominated by the deposition of gaseous Hg species or by the deposition of PBM. The source of Hg in wet deposition includes in-cloud oxidation of GEM to soluble Hg(II) and scavenging of GOM and PBM out of the atmosphere.

Rates of dry deposition are commonly calculated as the product of a measured Hg atmospheric concentration and an assumed deposition velocity [11]. The magnitude of deposition velocity is relatively small for GEM (0.1–0.4 cm/s), moderate for PBM (0.02–2 cm/s), and large for GOM (0.5–6 cm/s). An alternative method to estimate Hg dry deposition is a direct approach that captures deposited Hg onto a variety of abiotic surfaces including water surfaces, membranes and artificial turf [12–15]. This method has the advantage of obviating the need to assume a deposition velocity. However, uncertainty exists as to how well surrogate surfaces mimic environmental surfaces. In this study, we used a direct wet-sampler to measure dry deposition in the summer and fall of 2011 to compare patterns at a rural site in eastern Washington and a suburban site in western Washington. Wet deposition was also assessed with a simple funnel and Teflon sampling bottle. Since urban locations tend to have higher rates of Hg deposition, we anticipated that the urban site would exhibit higher rates of Hg deposition. However, as illustrated in this study and acknowledged in other studies, this linkage can be weak since Hg deposition is affected not only by local emissions but also atmospheric processes [1,16]. We also assessed the significance of ephemeral summer storms on the annual Hg deposition budget for the rural site in eastern Washington and compared directly measured Hg deposition to results from a regional air quality model (AIRPACT-3).

This project was motivated by calls for expanded efforts to complement indirect measurement of Hg deposition with direct measurements that assess both dry and wet Hg deposition using inexpensive and simple approaches [14,17–19]. It was also motivated by the limited amount of Hg deposition data in Washington State and the greater Pacific Northwest [20–22]. While the data presented here is from a study several years ago, we were prompted to publish it because of its good fit with this special issue of Atmosphere focusing on atmospheric mercury monitoring. We believe our study’s results are still relevant and highlight the utility of direct monitoring of mercury deposition.

2. Experiments
2.1. Sampling Sites

Hg deposition was measured at two contrasting sites in Washington State (Figure 1). One site was located in Pullman in rural eastern Washington (46.733° N, 117.172° W). Samples were collected on the rooftop of Dana Hall, a three-story building (9.8 m height) on the Washington State University (WSU) campus. Pullman is surrounded by wheat, barley, lentil, and pea agricultural fields. The regional climate is semi-arid with hot, dry summers and cold, wet winters. Annual precipitation is ~50 cm/y. Of potential
importance to this study, agricultural field burning in the region is common in the fall, mainly during August and September. Dry deposition samples were collected in Pullman during four precipitation-free sampling events in August–September 2011. Wet deposition was measured during four events in June, July and October 2011. Rainfall events ranged in duration from 2 to 21 h and had mean precipitation intensity of 0.08–0.61 mm/h.

The other site was located in suburban Puyallup in western Washington (47.185° N, 122.292° W). Samples were collected on the rooftop of the Avian Health and Food Safety Laboratory (4 m height) at the WSU Puyallup Research and Extension Center. The site is southeast of the heavily urbanized and industrialized Seattle–Tacoma metropolitan area. The Centralia Power Plant, which is 73 km southwest of the study site, is the only coal fire power plant in Washington and is a major source of Hg deposition to western Washington. Puyallup is also relatively vulnerable to Hg deposition from long-range atmospheric transport of Hg from Asia [9]. The regional climate of Puyallup is cool with dry summers and mild, wet, and cloudy winters. Annual precipitation is ~100 cm/y. Dry deposition samples were collected during six sampling events in September–October 2011. Wet deposition was measured during three events in September–October 2011. Rainfall events ranged in duration from 16 to 56 h and had mean precipitation intensity of 0.30–0.38 mm/h.

2.2. Dry Deposition Sampling

The dry deposition sampling apparatus was based on [23,24]. A wet sampler was used that collected dry deposition in an acidic aqueous solution that was analyzed for total Hg (THg). The wet sampler was anticipated to mainly collect deposited GOM and PBM, but also some GEM since the solubility of elemental Hg is enhanced in low-pH water [25]. The sampler consisted of a Teflon plate, 35 cm in diameter, holding a thin layer (2.5 mm) of aqueous receiving solution (Figure 2). The plate was held by a collector with an outer edge shaped like an airfoil to minimize air flow disturbance over the water surface. The receiving solution was continuously circulated onto the plate via a pump and Teflon tubing. Solution was discharge onto the top of the plate then flowed over four weirs on the outer edge of the plate and into the collector, which was connected to a 5 L glass reservoir. The reservoir was chilled by a refrigeration unit to 10 °C to minimize evaporative losses.
Before each sampling event, the water surface holder, plate, tubing, and reservoir were cleaned thoroughly with 25% nitric acid and triple rinsed with reagent grade deionized water. The reservoir was filled with 5 L of 0.05 M hydrochloric acid (HCl) receiving solution which was circulated through the system at a rate of 200 mL/min. After 30 min of operation, an initial sample of receiving solution was collected for later Hg analysis. The system was operated for several days and a second sample of receiving solution was collected at the end of the sampling event. To determine the final volume, the remaining receiving solution was decanted and weighed. In the first Pullman and all Puyallup sampling events, two samples were collected, one mid-way through the sampling event and another at the end of the sampling event.

Samples were preserved with 1% bromine monochloride and analyzed in triplicate for THg based on United States Environmental Protection Agency (USEPA) Method 1631 [26] on a Brooks Rand MERX-T mercury auto analyzer. The method detection limit for THg is 0.2 ng/L. Standard quality control procedures for THg analyses included calibration blanks (acceptable range < 50 pg; mean = 2.98 pg, n = 20), matrix spike samples (acceptable range 71–125%; mean recovery = 88.6%, n = 15), and ongoing precision (acceptable range 77–123%; mean recovery = 96.6%, n = 24). Mass accumulation of THg was estimated as the difference between the concentration multiplied by associated volume of receiving solution at the end and beginning of the sampling event. Dry Hg deposition in ng/m²·hr was calculated as the mass accumulation divided by the area of the plate (0.0962 m²) and the duration of the sampling event.

2.3. Wet Deposition Sampling

Wet deposition sampling apparatus was based on [27]. The sampler included four components: a borosilicate glass funnel with a 20-cm-diameter opening, a Teflon adaptor, a 1 L Teflon sampling bottle, and a Plexiglas holder (Figure 2). Prior to each sampling event, the borosilicate glass funnel, adaptor, and sampling bottle were cleaned thoroughly with 25% nitric acid and triple rinsed with reagent grade deionized water. Twenty ml of 0.08 M HCl receiving solution was added to the sample bottle to enhance capture and preservation of Hg in collected precipitation. A subsample of receiving solution was collected at the beginning of the sampling event for later THg analysis. At the end of the sampling event, collected rainwater was weighed to determine the volume of precipitation, then a sample was collected. Samples were preserved and analyzed for THg as described above for the dry deposition monitoring, with the exception that Pullman June and July samples were not run in triplicate due to limited sample volume. Mass accumulation of Hg was estimated as the THg concentration in the rainfall-receiving solution mixture collected at the end of the sampling event multiplied by its volume, corrected for the initial THg mass in the receiving solution. Wet Hg deposition in ng/m²·hr was calculated as mass accumulation divided by the area of the funnel (0.0314 m²) and the duration of the sampling event.
3. Results

3.1. Dry Deposition

Dry Hg deposition measured in Pullman ranged from 1.0 to 4.3 ng/m²·h and averaged 2.4 ng/m²·h (Table 1, Figure 3). The high deposition rate observed in late August (4.3 ng/m²·h) corresponded with smoky conditions in Pullman. High air temperatures (12–26 °C), low relative humidity (31–54%), and elevated wind speeds (0.8–3.1 m/s) during August sampling resulted in high rates of evaporation [28]. Normalized to the area of the sampler, evaporation rates (5.4–5.9 mm/d) were similar to the historical mean August pan evaporation rate for the area (6.7 mm/d) [29]. Dry Hg deposition in Puyallup ranged from 0.8 to 1.6 ng/m²·h and averaged 1.3 ng/m²·h. Relative to Pullman, dry deposition in Puyallup was lower in magnitude and less variable. Measured evaporation rates were relatively low and corresponded with low air temperature (11–22 °C), high relative humidity (66–88%), and low wind speed (0–0.6 m/s) during September sampling [28]. Estimated evaporation rates for the three events (0.3–2.0 mm/d) were lower than the historical mean September pan evaporation rate for the region (2.4 mm/d) [29].

| Date          | Duration (Days) | Initial Vol (L) | Final Vol (L) | Evap Rate (L/d) | Initial Total Hg a (ng/L) | Final Total Hg a (ng/L) | Hg Dry Dep b (ng/m²·h) |
|---------------|-----------------|-----------------|---------------|-----------------|---------------------------|-------------------------|------------------------|
| Pullman, WA   |                 |                 |               |                 |                           |                         |                        |
| 15–18 Aug     | 2.77            | 4.87            | 3.28 c        | 0.57 d          | 1.57                      | 7.12                    | 2.5                    |
| 18–22 Aug     | 3.96            | 3.02            | 0.75          | 7.12            | 41.1                      | 28.9                    | 4.3                    |
| 23–28 Aug     | 5.04            | 4.74            | 1.96          | 0.03 d          | 2.37                      | 3.45                    | 1.2                    |
| 29 Aug–5 Sep  | 7.11            | 4.74            | 1.02          | 0.16 d          | 1.37                      | 4.01                    | 1.5                    |
| Puyallup, WA  |                 |                 |               |                 |                           |                         |                        |
| 12–15 Sep     | 3.03            | 4.75            | 4.26 c        | 0.19 d          | 2.21                      | 4.48                    | 1.0                    |
| 15–17 Sep     | 1.64            | 4.01            | 3.74          | 2.37            | 3.45                      | 1.2                    |
| 21–23 Sep     | 2.05            | 4.75            | 4.69 c        | 2.21            | 4.48                      | 1.0                    |
| 23–25 Sep     | 1.74            | 4.43            | 4.38          | 2.37            | 3.45                      | 1.2                    |
| 27–29 Sep     | 2.00            | 4.75            | 4.37 c        | 2.21            | 4.48                      | 1.0                    |
| 29 Sep–2 Oct  | 2.97            | 4.11            | 3.55          | 2.21            | 4.48                      | 1.0                    |

Table 1. Mercury dry deposition.

* Average of triplicate analysis. b Deposition based on sampler cross-sectional area of 0.096 m². c Volume estimated based on evaporation rate over entire event. d Evaporation rate based on total volume loss over entire event.

Figure 3. Dry and wet Hg deposition at Pullman and Puyallup, WA. Values are mean plus one standard deviation (n = 3–6).
3.2. Wet Deposition

Wet Hg deposition measured in Pullman ranged from 1.8 to 12.8 ng/m$^2$·h and averaged 7.0 ng/m$^2$·d (Table 2, Figure 3). Deposition rates for the three storm events in June/July were an order of magnitude higher than for the October event. Two summer events in June and July with low precipitation intensity (<0.17 mm/h) had relatively high mean Hg concentrations in precipitation (>56 ng/L). Wet Hg deposition in Puyallup in September/October ranged from 0.9 to 1.3 ng/m$^2$·h. Mean Hg concentrations in Puyallup precipitation ranged from 2.3 to 3.5 ng/L. As with dry deposition, wet deposition rates and Hg concentrations in precipitation were lower and less variable in Puyallup when compared to Pullman.

Table 2. Mercury wet deposition.

| Date       | Duration (h) | Initial Vol (mL) | Final Vol (mL) | Precip (mm/h) | Initial Total Hg a (ng/L) | Final Total Hg a (ng/L) | Hg Wet Dep c (ng/m$^2$·h) | Precip Hg (ng/L) |
|------------|--------------|------------------|----------------|---------------|---------------------------|------------------------|------------------------|------------------|
| Pullman, WA| 28–29 June   | 3.50             | 20             | 38.9          | 0.17                       | 0.81                   | 36.1 b                 | 12.8             |
|            | 12–13 July   | 11.0             | 20             | 48.5          | 0.08                       | 0.90                   | 33.8 b                 | 4.8              |
|            | 14 July      | 2.17             | 20             | 61.2          | 0.61                       | 0.88                   | 9.84 b                 | 8.7              |
|            | 10–11 Oct    | 21.0             | 20             | 210           | 0.29                       | 0.27                   | 5.74                   | 1.8              |
| Puyallup, WA| 17–19 Sep    | 37.9             | 20             | 467           | 0.38                       | 0.71                   | 2.27                   | 0.89             |
|            | 25–27 Sep    | 55.4             | 20             | 533           | 0.30                       | 0.86                   | 3.28                   | 1.0              |
|            | 2–3 Oct      | 16.1             | 20             | 209           | 0.38                       | 0.62                   | 3.18                   | 1.3              |

a Average of triplicate analysis. b Single analysis due to low precipitation volume. c Deposition based on funnel cross sectional area of 0.031 m$^2$.

4. Discussion

4.1. Dry Deposition

Dry Hg deposition rates observed in this study, typically 1–2 ng/m$^2$·h, were similar to a limited number of studies that report direct measurements of dry Hg deposition using wet samplers with acidified de-ionized water as receiving solution. Dry Hg deposition measured in Komae, a heavily industrialized city in the western Tokyo metropolitan area, ranged from 0.5 to 3 ng/m$^2$·h ($n = 5$) [24]. Related long-term studies measured Hg deposition monthly at 10 sites in Japan over multiple years. Mean dry deposition ranged from 0.4–2 ng/m$^2$·h and higher deposition rates were associated with urban and industrial areas [30,31]. Dry deposition ranging from 0.4 to 1.7 ng/m$^2$·h ($n = 5$) was measured in rural New York State [32]. Some studies have used un-acidified de-ionized water as receiving water in passive static samplers and report dry deposition rates ranging from 0.5 to 2 ng/m$^2$·h in an urban site in Ohio [14] and 0.3–0.5 ng/m$^2$·h in the Florida Everglades [33]. Measured deposition rates generally increase with the addition of acid, presumably due to enhanced capture and retention of GEM [25].

In an effort to compare our direct measurements of dry deposition with those estimated based on atmospheric Hg monitoring, we searched the USEPA AirData Air Quality Monitors website and found relevant data in the National Air Toxics Trends Stations (NATTS) database [34]. While no stations were located in the rural eastern portion of Washington, a station was located near our Puyallup site (site code 53/033/0080; 47.56824° N, 122.309° W). Reported arithmetic mean daily values for Hg PM10 concentrations for 2011 ranged from 0 to 0.04 ng/m$^3$ over the year ($n = 60$) and were ~0.01 ng/m$^3$ during this study (Table S1). Back calculating a depositional velocity during the study period as the mean measured dry deposition flux (1.25 ng/m$^2$·h; Table 1) divided by the mean measured Hg PM10 (0.0075 ng/m$^3$; Table S1) yields a deposition velocity on the order of 2.6 cm/s,
which is similar in magnitude but on the high end of deposition velocity typically reported for PBM (0.02–2 cm/s) [11].

The rural site in Pullman exhibited rates of dry Hg deposition that were somewhat higher in magnitude when compared to the suburban Puyallup site. While rural, the landscape around Pullman is intensely managed for agricultural production. When crops are harvested, Hg deposited on soil and plants during the growing season can be dispersed into the atmosphere on dust and deposited back onto the landscape, resulting in enhanced localized Hg deposition [35]. Agricultural field tilling has also been implicated in elevated levels of atmospheric Hg and downwind Hg deposition [36]. Wildfires and agricultural fires can also substantially enhance Hg emissions via combustion of biomass, litter and organic soils, with the magnitude of emissions apparently corresponding with fire severity [10]. A 3–4 fold increase in total gaseous Hg was measured above wild and agricultural fires in eastern Washington and Oregon [37]. Hg emitted via fires can enhance regional Hg deposition rates [3]. A 12-fold increase in PBM deposition was measured in New Mexico resulting from transport of smoke from a large forest fire in Arizona [38].

A review of regional burn permits, along with regional meteorology and air quality, suggests that the relatively elevated dry Hg deposition in Pullman measured between 15 August and 5 September was affected by agricultural field burning. Burn permits allotted by the Washington State Department of Ecology indicated that agricultural field burns totaling 400 ha occurred in Walla Walla County, 160 km southwest of Pullman, between 15 and 27 August. Prevailing wind direction in the region during August 2011 was generally southwesterly. Mean hourly wind direction from 15 to 27 August was 236° ± 68° (mean ± standard deviation, n = 142) in Pullman (http://mesowest.utah.edu; station KPUW) and 183° ± 70° (n = 266) in Walla Walla (http://mesowest.utah.edu; station KALW). Back trajectory modeling using HYSPLIT, a hybrid single-particle Lagrangian integrated trajectory model hosted by the Nation Oceanic and Atmospheric Administration (http://www.arl.noaa.gov/HYSPLIT.php), confirmed that air parcels in Pullman generally passed through the Walla Walla region throughout mid to late August 2011 (Figure S1). In addition, because wildfires and prescribed fires are known to be substantial sources of atmospheric particulate matter [39], we assessed PM2.5 concentrations reported via the USEPA Outdoor Air Quality Data repository (https://www.epa.gov/outdoor-air-quality-data/download-daily-data) for Pullman (station 530710005) and Walla Walla (station 530750003). PM2.5 concentrations were generally <3 μg/m³ in July and <5 μg/m³ in August, with levels increasing to >10 μg/m³ at times in late August and early September (Figure S2), suggesting that smoky conditions were present in Pullman when dry deposition measurements were performed in this study. Relatively high levels of PM2.5 in Pullman between 23 and 28 August (Figure S2) corresponded with the highest measured Hg dry deposition of 4.3 ng/m²·h (Table 1), further suggesting a linkage between smoky conditions and Hg deposition.

4.2. Wet Deposition

Except for the June and July events in Pullman, which had elevated THg concentrations in precipitation of 73 ng/L and 57 ng/L, respectively, the results of this study were comparable to concentrations reported worldwide, which typically range from 1 to 15 ng/L [40]. Concentrations are also similar to those measured in California precipitation, which ranged from 1 to 28 ng/L and averaged 4 ng/L (n = 46) [41]. The extreme concentrations measured in Pullman appear to be relatively uncommon at other reported monitoring sites. In hundreds of precipitation samples from Europe and China, 95th percentile THg concentrations rarely exceeded 40 ng/L [40]. The high Pullman THg concentrations were undoubtedly related to the small precipitation amount associate with these monitored events. While annual Hg deposition generally correlates with total annual precipitation, THg concentration in collected precipitation generally shows a negative correlation with precipitation amount [1]. This observation is attributed to scavenging of Hg from the air column early during storm events and subsequent dilution of the accumulating sample
later in the event. The relationship between THg concentration and precipitation amount was apparent in our pooled samples (Figure 4). This phenomenon, combined with the fact that agricultural activities can enhance Hg re-emission from the landscape [35,36], explains the high Hg concentrations in precipitation observed during small rain events in Pullman in June and July.

![Figure 4. Relationship between Hg concentration in precipitation and precipitation depth for data for Pullman and Puyallup, WA. Line is linear regression.](image)

While brief in duration, wet Hg deposition via ephemeral storm events observed in Pullman in June and July deposited a similar magnitude of Hg to longer storm events. For example, the 3.5-h-long June storm event deposited 45 ng/m\(^2\) while the 21-h-long October storm event deposited 39 ng/m\(^2\). While the frequency of monitoring in this study was admittedly limited, the data obtained can be used to assess patterns of Hg deposition during the dry period in Pullman (May–September 2011). Based on an examination of 2011 precipitation patterns for the dry period in Pullman and mass balance considerations, ephemeral storm events may have accounted for ~20% of dry-season Hg deposition (see Section S1 in Supplemental Material for additional details).

4.3. Model Versus Direct Measurements of Hg Deposition

AIRPACT-3 (Air Indicator Report for Public Access and Community Tracking) is a numerical air quality forecast system that operates daily for the Pacific Northwest. It estimates a range of air quality parameters including Hg dry and wet deposition (see Section S2 in Supplemental Material for additional details). The modeled rates of Hg deposition from AIRPACT-3 were generally lower, but of similar magnitude, compared to direct measurements in the field. For Pullman (Figure S3), modeled dry deposition typically ranged from 1 to 5 ng/m\(^2\)-d with occasional peaks of 20–40 ng/m\(^2\)-d. Modeled peaks of dry Hg deposition were similar in magnitude but shorter in duration than direct measurements in mid-August (37 ng/m\(^2\)-d) and early September (39 ng/m\(^2\)-d). Modeled values were far below the extreme dry deposition rate of 103 ng/m\(^2\)-d measured in late August, likely associated with enhanced deposition from regional agricultural field burning. Modeled rates of wet deposition showed four events with substantial deposition (>5 ng/m\(^2\)-d), one in June, one in July, and two in October. The July and October modeled deposition events coincided temporally with direct measurement field events, indicating that the model captured actual
precipitation events, including occasional ephemeral storm events during the summer. However, direct measurements on those two dates were 5–10 times the model estimates.

For Puyallup, modeled dry deposition typically ranged from 3 to 15 ng/m$^2$·d with occasional peaks of 30–60 ng/m$^2$·d (Figure S4). Measured dry deposition was higher, ranging from 28 to 33 ng/m$^2$·d. The model captured the temporal dynamics of precipitation events associated with direct field measurements fairly well, including dry conditions during all three dry deposition monitoring events and wet conditions during the second and third wet deposition monitoring events, though the timing of the third event in early October was a bit early relative to observed precipitation in Puyallup. Based on the last two monitoring events in late September and early October, direct measurements of wet deposition were 2–4 times higher than those predicted by the model. Considering that this study compared a limited number of direct measurements at discrete points in space to modeling results on a 12-km grid, rates of Hg deposition predicted by AIRPACT-3 were reasonably close to direct measurements.

5. Conclusions

This study used a direct approach to assess the magnitude of dry and wet Hg deposition at a rural and suburban site in Washington. Measured values, generally ranging from 1 to 2 ng/m$^2$·h, are some of the first published direct measurements of Hg deposition in the State. Key conclusions of the study include:

1. Because of their relative simplicity and robustness, direct measurement approaches such as those described in this study are useful in assessing temporal and spatial patterns of Hg deposition, and for comparing results to less direct estimates of Hg deposition and estimates from numerical air quality models.

2. Hg deposition can be substantial in rural regions with significant agricultural activities. Hg deposition rates at the rural study site (Pullman, Washington) were similar to or higher than deposition rates observed at the suburban study site (Puyallup, Washington), which was likely influenced by regional urban and industrial sources of Hg.

3. In rural agricultural areas, agricultural burning and associated re-emission and transport of previously deposited Hg can lead to elevated levels of Hg dry deposition. Rates of dry deposition in Pullman during smoky conditions indicative of agricultural burning were ~2.5 times the deposition rates observed during non-smoky conditions.

4. Hg concentrations in precipitation correlated negatively with precipitation depth, suggesting that scavenging of PBM and GOM from the atmosphere at the beginning of storm events was an important wet deposition process.

5. Ephemeral, short-term storm events at the rural Pullman site had elevated Hg concentrations. Mass balance estimates indicated that these Hg-rich storm events may account for a meaningful fraction (~20%) of dry season Hg deposition.

Supplementary Materials: The following are available online at https://www.mdpi.com/2073-4334/12/1/35/s1: Table S1: Atmospheric Hg PM10 concentrations measured near Puyallup, WA; Figure S1: Example HYSPLIT back trajectories from Pullman, WA; Figure S2: PM2.5 data for Pullman and Walla Walla, WA; Section S1. Ephemeral Storm Events as Source of Hg Deposition; Section S2. AIRPACT-3 Air Quality Model; Figure S3: Measured and modeled dry and wet Hg deposition for Pullman, WA; Figure S4: Measured and modeled dry and wet Hg deposition for Puyallup, WA.

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