The rare cis-configured trisubstituted lactam products obtained by the Castagnoli–Cushman reaction in N,N-dimethylformamide†

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Unlike its trans counterpart, the cis-configured scaffold derived from the Castagnoli–Cushman reaction (CCR) is scarce and has not been explored in bioactive compound design. We found that conducting this reaction in DMF (in contrast to conventional toluene or xylene) led to a significantly higher proportion of cis-configured lactams in the diastereomeric product mixture. This allowed us, for the first time, to obtain and thoroughly characterize both stereoisomers of a significant number of lead-like CCR lactams. Simple rules for 1H NMR-based stereochemical assignment have been devised and correlated with the single-crystal X-ray structures obtained for pure cis- and trans-configured lactams.

Discovered in 1969,† the Castagnoli–Cushman reaction (CCR) of imines 1 (formed in situ or in a separate step), succinic (2) or glutaric (3) anhydrides provides convenient access to the medicinally important polysubstituted lactams 4–5 in a convenient, three-component format (Scheme 1).

CCR has recently emerged‡ as an important synthetic tool for lead-oriented synthesis (LOS).§ Considering the challenges of inventing new LOS methods‖ and our recent interest in developing lead-like compound libraries,‖ particularly those employing the CCR,‡ we thought it important to try and overcome certain limitations of the CCR, which were evident from the literature. Besides limitations of scope which we discuss elsewhere,‖ several drawbacks associated with CCR are noteworthy. Firstly, in a vast majority of cases reported in the literature, the reaction involving simple anhydride inputs (such as toluene or xylene) and requires rather forcing conditions (reflux). These two aspects introduce solubility and structural tolerability limitations, respectively.‡ Secondly, almost universally the reaction was reported as proceeding with high diastereoselectivity (up to 95 : 5) and only the major, trans-isomer was isolated and characterized (with the exception of the pioneering reports by Castagnoli† the minor, cis-isomer was isolated in <1% yield by fractional crystallization). Indirect synthesis of the cis counterpart of 4 (compound 6) via desulfurization (with reversal of configuration) of tetrasubstituted lactams 7 (obtainable, in turn, via four-component reaction of maleic anhydride) was reported by Shaw (Scheme 2).§ However, the cis-configured stereoisomer of 5 remain virtually unavailable. Multistep sequences leading to the cis-isomer of related δ-lactams have been reported by Stille‖ and others.‖ However, these strategies lack the practicality and flexibility offered by the multicomponent approach. From medicinal chemistry perspective, cis- and trans-versions of either 4 or 5 represent distinct scaffolds‡‡ that ensure different spatial projection of the molecular periphery and, therefore, are likely to have vastly different complementarities to a protein target. The fact that the cis-isomer is always formed in impractical minority severely curbs the stereochemical diversity‡‡ attainable by the CCR and represents a significant limitation of this chemistry.

In this work, we sought to find a solution to the above-mentioned limitations by drastically changing the solvent in which the CCR would be conducted. While toluene and xylene indeed appear to be solvents of choice (from the literature review), we ventured to investigate the CCR of more...
than two dozen three-component reactions in $N,N$-dimethylformamide (DMF). The choice of high-boiling DMF as a reaction medium in lieu of toluene or xylene was dictated by its higher polarity and ability to dissolve a wider range of reactants. In this Communication, we report the results of these studies.

Table 1  $\gamma$- and $\delta$-lactams 4 and 5 prepared in this work via the CCR in DMF

| Compound | $R^1$       | $R^2$ | n | Isolated yield (%) | Reaction time | dr (cis : trans) |
|----------|-------------|-------|---|--------------------|---------------|-------------------|
| 4a       | 4-MeOC$_2$H$_4$ | Bn    | 1 | 74                 | 5 h           | 1 : 3.5           |
| 4b       | 4-MeOC$_2$H$_4$ | i-Pr  | 1 | 75                 | 5 h           | 1 : 1.8           |
| 4c       | 4-MeOC$_2$H$_4$ | t-Bu  | 1 | 69                 | 5 h           | 1 : 1.7           |
| 4d       | 4-FC$_6$H$_4$ | n-Pr  | 1 | 46                 | 5 h           | 1 : 2.5           |
| 4e       | 4-MeC$_6$H$_4$ | Et    | 1 | 70                 | 5 h           | 1 : 2             |
| 4f       | 4-BrC$_6$H$_4$ | Bn    | 1 | 48                 | 24 h          | 1 : 3             |
| 4g       | 4-O$_2$NC$_6$H$_4$ | n-Bu | 1 | 28                 | 24 h          | 1 : 4.5           |
| 4h       | 3-Py        | n-Pr  | 1 | 62                 | 6 h           | 1 : 2             |
| 4i       | 3-O$_2$NC$_6$H$_4$ | Me | 1 | 48                 | 6 h           | 1 : 1.8           |
| 4j       | 4-Me$_2$NC$_6$H$_4$ | Allyl | 1 | 84                 | 6 h           | 1 : 2             |
| 4k       | 1-Naphthyl | n-Pr  | 1 | 59                 | 6 h           | 1 : 2.1           |
| 5a       | 2-Thienyl | cyclo-Pr | 2 | 35                 | 7 days        | 1 : 2.3           |
| 5b       | 4-MeOC$_2$H$_4$ | Et    | 2 | 53                 | 6 h           | 1 : 2.2           |
| 5c       | 4-Me$_2$NC$_6$H$_4$ | n-Bu | 2 | 62                 | 6 h           | 1 : 2             |
| 5d       | 4-MeC$_6$H$_4$ | n-Pr  | 2 | 21                 | 6 h           | 1 : 2.2           |
| 5e       | 3-Py       | n-Pr  | 2 | 16                 | 24 h          | 1 : 2.2           |
| 5f       | 4-MeC$_6$H$_4$ | Bn    | 2 | 52                 | 24 h          | 1 : 1.7           |
| 5g       | 4-BrC$_6$H$_4$ | n-Pr  | 2 | 58                 | 24 h          | 1 : 3.6           |
| 5j       | 4-AcNH$_2$C$_6$H$_4$ | cyclo-Pr | 2 | 54                 | 48 h          | 1 : 2.3           |
| 5l       | 3-Py       | n-Pr  | 2 | 61                 | 24 h          | 1 : 2.5           |
| 5m       | 4-MeOC$_2$H$_4$ | Ph   | 2 | 39                 | $22 h^a$      | 1 : 5             |
| 5n       | 4-MeC$_6$H$_4$ | 4-MeOC$_2$H$_4$ | 2 | 47                 | $22 h^a$      | 1 : 4.2           |
| 5o       | 4-MeC$_6$H$_4$ | 4-MeOC$_2$H$_4$ | 2 | 53                 | $22 h^a$      | 1 : 4.1           |

$^a$ Reaction temperature was raised to 125 °C in order to ensure maximum conversion.
A brief screening of various temperature regimens for the reactions leading to 4a and 5a (R^2 = 4-MeOC_6H_4, R^2 = Bn) revealed that changing the solvent medium from toluene to DMF offered little advantage in terms of lowering the reaction temperature. However, it was immediately evident that the formation of the cis-isomer was significantly more favored in DMF compared to toluene (Table S1 in ESI†). Particularly encouraging were the dr values obtained with glutaric anhydride at 110 °C (cis : trans = 1 : 2.5), considering the fact that, apart from the initial reports by Castagnoli and Cushman (vide supra),^1 cis-configured δ-lactams 5 were practically unattainable by the CCR. It should be noted that, occasionally, other solvents (such as DMSO) also favored the formation of the cis-configured products (see ESI†). However, this can more reliably achieved with DMF as a solvent of choice.

Encouraged by these results, we extended the new reaction protocol to a range of Schiff bases (pre-formed by reaction of a primary amine with an aldehyde in presence of MgSO4 or molecular sieves) which underwent CCR with succinic (1) and glutaric (2) anhydrides at 110 °C in DMF over 5–24 h period (Scheme 3).

Table 2 Comparative summary of the 1H NMR spectral characteristics of cis- and trans-isomers for compounds 4 and 5

| Compound | 3J (C=H–H), Hz | δ (C=H), ppm (multiplicity) | δ (C=H), ppm (multiplicity) | Δ [δ(C=H)] cis/trans, ppm |
|----------|----------------|-----------------------------|-----------------------------|--------------------------|
| cis      | trans          | cis                        | trans                      |                          |
| 4a       | 9.0            | 5.8                         | 4.64 (d)                   | 4.50 (d)                 | 3.54 (dt)                 | 3.06 (ddd)                 | 0.48                      |
| 4b       | 9.1^a          | 4.4^a                       | 4.97 (d)                   | 4.79 (d)                 | 3.61 (dt)                 | 2.89 (ddd)                 | 0.72                      |
| 4c       | 8.6^a          | 1.0                         | 5.13 (d)                   | 5.09 (d)                 | 3.53 (dt)                 | 2.62 (ddd)                 | 0.91                      |
| 4d       | 9.2^a          | 5.6^a                       | 5.00 (d)                   | 4.81 (d)                 | 3.66 (dt)                 | 3.02 (ddd)                 | 0.64                      |
| 4e       | 9.3^a          | 5.9                         | 4.99 (d)                   | 4.76 (d)                 | 3.63 (dt)                 | 3.00 (ddd)                 | 0.63                      |
| 4f       | 9.3            | 5.5                         | 4.72 (d)                   | 4.55 (d)                 | 3.66 (dt)                 | 3.07 (ddd)                 | 0.59                      |
| 4g       | 9.2            | 5.4                         | 5.17 (d)                   | 4.99 (d)                 | 3.72 (dt)                 | 3.06 (ddd)                 | 0.66                      |
| 4h       | 9.3            | 5.7                         | 5.06 (d)                   | 4.85 (d)                 | 3.73 (dt)                 | 3.11 (ddd)                 | 0.62                      |
| 4i       | 9.3            | 6.2                         | 5.13 (d)                   | 4.94 (d)                 | 3.71 (dt)                 | 3.11 (ddd)                 | 0.60                      |
| 4j       | 9.0            | 5.7                         | 4.77 (d)                   | 4.62 (d)                 | 3.64 (dt)                 | 3.01 (ddd)                 | 0.63                      |
| 5a       | 5.0            | 5.6                         | 4.71 (d)                   | 4.70 (d)                 | 3.11 (ddd)                 | 2.78 (m)                  | 0.33                      |
| 5b       | 4.6^a          | 2.8                         | 4.93 (d)                   | 4.98 (d)                 | 3.00 (dt)                 | 2.72 (m)                  | 0.29                      |
| 5c       | 4.1^a          | 2.1                         | 5.22 (d)                   | 5.38 (d)                 | 3.01 (dt)                 | 2.79 (dt)                 | 0.30                      |
| 5d       | 5.1            | 4.1                         | 4.95 (d)                   | 4.97 (d)                 | 3.18 (m)                  | 2.82 (dt)                 | 0.36                      |
| 5e       | 4.7            | 3.6                         | 5.17 (d)                   | 5.19 (d)                 | 3.25 (m)                  | 2.95 (m)                  | 0.18                      |
| 5f       | 4.5^a          | 4.2                         | 4.91 (d)                   | 4.90 (d)                 | 3.12 (m)                  | 2.79 (dt)                 | 0.33                      |
| 5g       | 4.6            | 4.1                         | 4.88 (d)                   | 4.89 (d)                 | 3.10 (m)                  | 2.85 (dt)                 | 0.25                      |
| 5h       | 5.2            | 3.7                         | 5.18 (d)                   | 5.15 (d)                 | 3.16 (m)                  | 2.88 (m)                  | 0.27                      |
| 5i       | 4.9            | 4.0                         | 5.10 (d)                   | 5.17 (d)                 | 3.04 (ddd)                 | 2.81 (dt)                 | 0.23                      |
| 5j       | 5.4            | 2.3                         | 5.91 (d)                   | 5.85 (br s)              | 3.33 (m)                  | 2.98 (dt)                 | 0.35                      |
| 5k       | 4.9            | 4.3                         | 4.84 (d)                   | 4.87 (d)                 | 3.08 (m)                  | 2.78 (dt)                 | 0.37                      |
| 5l       | 5.2            | 4.2                         | 5.20 (d)                   | 5.04 (d)                 | 3.26 (ddd)                 | 2.89 (dt)                 | 0.30                      |
| 5m       | 5.0            | 4.2                         | 5.29 (d)                   | 5.39 (d)                 | 3.49 (ddd)                 | 2.98 (dt)                 | 0.51                      |
| 5n       | 4.9^a          | 4.2                         | 5.17 (d)                   | 5.26 (d)                 | 3.45 (ddd)                 | 2.91 (dt)                 | 0.54                      |
| 5o       | 5.1            | 4.5                         | 5.16 (d)                   | 5.23 (d)                 | 3.44 (ddd)                 | 2.91 (dt)                 | 0.53                      |

^a Stereochemical assignment confirmed by single-crystal X-ray analysis.
With a strong base, such as potassium tert-butoxide, prolonged reaction times can cause a mixture. This, however, clearly indicates that, in order to maximize the formation of the CCR products, the reaction temperature was raised to 125 °C and the reaction time extended to 22–24 h, in order to reach maximum conversion. This, in turn, was associated with lower proportion of the cis-configured γ-lactam in the product mixture.

Considering the latter observation, we were curious to see if prolonged reaction times can cause a ‘drift’ in the observed dr values, via a gradual isomerization of the cis-isomer into the thermodynamically more stable trans-isomer (a transformation certainly achievable with good yield on treatment with a strong base, such as potassium tert-butoxide). To verify that, diastereomeric mixtures of 4a (cis/trans = 1 : 3.5) and 5a (cis/trans = 1 : 2) were heated in DMF at 110 °C for a period of up to 10 days with periodic monitoring of the dr. As the result, the dr of 4a changed drastically to 1 : 13.8 while 5a was somewhat more resistant to thermal isomerization: its dr only changed to 1 : 3.9. This, however, clearly indicates that, in order to maximize the formation of the cis-isomer, the CCR should be closely monitored and the reactions should be stopped as soon as the maximum conversion is reached (the data reported in Table 1 was obtained in this manner).

Formation of a higher proportion of the cis-isomer of lactams 4 and 5 allowed, for the first time, separation, on preparative scale, and full characterization of cis- and trans-isomers of compounds presented in Table 1 (see ESI†). There has not been much progress in comparative characterization of diastereomeric CCR products since the pioneering work of Castagnoli and Cushman. Therefore, in this work we also aimed to establish, using the wide range of compounds synthesized, a basis for assigning cis- or trans-configuration to the γ- and δ-lactams obtainable by the CCR based on their NMR spectra and to correlate these findings with a single-crystal X-ray crystallography data.

As it is evident for the value of the coupling constants of the C2–H protons provide a rather solid basis for stereochemical assignment of γ-lactams 4a–j, as was also reported earlier. However, the difference in the same coupling constants is not that pronounced for δ-lactams 5a–o and varies from 0.5 to 2.0 Hz (though the value appears to be always larger for the cis-isomer compared to its trans-counterpart). In an attempt to look for more characteristic patterns that could additionally aid in the stereochemical assignment, we also looked at the chemical shifts of C3–H and C4–H protons. While the former displayed no characteristic pattern, the latter appear to be universally positioned downfield for the cis-isomer compared to the trans-isomer. This difference also appears to be more pronounced for of γ-lactams 4 compared to δ-lactams 5 (Table 2). In order to ultimately confirm our stereochemical assignment (which was initially based on the dr values observed and then confirmed by the 1H NMR patterns displayed by the respective stereoisomers, as discussed above), we obtained a number of single-crystal X-ray structures of individual cis- and trans-isomers of the CCR lactams in both 4 and 5 series (see ESI†, a representative pair is shown in Fig. 1). To the best of our knowledge, until today, crystallographic information on the cis-isomers of the CCR lactams (whihm scarcely available) has been lacking in the literature.

The individual cis- and trans-isomers reported in this Communication were obtained by preparative reverse-phase HPLC separation of diastereomeric mixtures of carboxylic

![Scheme 4](image-url)

**Scheme 4** Diastereomer separation of methyl esters obtained from compound 5d.
acid lactams 4 and 5 (which are difficult to separate by conventional column chromatography). Alternatively, the carboxylic acids can be converted to the respective methyl esters, in order to facilitate chromatographic separation on silica gel.* This strategy was successfully realized for compound 5d (Scheme 4).

In conclusion, we established that the Castagnoli–Cushman reaction in DMF leads to higher yields of the cis-configured γ- and δ-lactams. This enabled preparative isolation and characterization of this rare type of lead-like scaffolds in comparison with their well-described trans-configured counterparts. This, in turn, led to a more reliable basis for stereochemical assignment based on characteristic $^1$H NMR patterns displayed by the two isomeric series, which was correlated with a number of X-ray structures. Efforts to unveil the medicinal chemistry potential of the cis-isomeric CCR lactams are underway in our laboratories. The results will be reported in due course.

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