Abstract
The global growth of clean energy technology deployment will be followed by parallel growth in end-of-life (EOL) products, bringing both challenges and opportunities. Cumulatively, by 2050, estimates project 78 million tonnes of raw materials embodied in the mass of EOL photovoltaic (PV) modules, 12 billion tonnes of wind turbine blades, and by 2030, 11 million tonnes of lithium-ion batteries. Owing partly to concern that the projected growth of these technologies could become constrained by raw material availability, processes for recycling them at EOL continue to be developed. However, none of these technologies are typically designed with recycling in mind, and all of them present challenges to efficient recycling. This article synthesizes and extends design for recycling (DfR) principles based on a review of published industrial and academic best practices as well as consultation with experts in the field. Specific principles developed herein apply to crystalline-silicon PV modules, batteries like those used in electric vehicles, and wind turbine blades, while a set of broader principles applies to all three of these technologies and potentially others. These principles are meant to be useful for stakeholders—such as research and development managers, analysts, and policymakers—in informing and promoting decisions that facilitate DfR and, ultimately, increase recycling rates as a way to enhance the circularity of the clean energy economy. The article also discusses some commercial implications of DfR.
Keywords  DfR · Circular economy · Renewable energy · Photovoltaics · PV · Batteries · Wind

Introduction

In the traditional linear economy, products are typically landfilled at end of life (EOL), and new products from virgin materials are manufactured to replace them. In contrast, a circular economy optimizes energy and material use over product life cycles by remanufacturing, refurbishing, repairing, or reusing EOL products. When products can no longer be remanufactured, refurbished, repaired, or reused, recycling is the final circular option. Yet many products comprise a complex mix of materials that are difficult to recover during recycling. One approach to bringing such products into the circular economy is to design them from the beginning with recycling in mind. Design for recycling (DfR) holds potential to increase the quantity and value of materials recovered and reused from EOL products.

Researchers have proposed engineering practices that resemble DfR under a variety of other names and classifications [1–5]. For example, the classification by Fiksel [2] identifies common pathways and methods among different design for the environment (DfE) strategies, where DfE encompasses DfR under the category of design for revalorization; other DfE categories include design for dematerialization, design for detoxification, and design for capital protection and renewal.

DfR is optimally applied when looking beyond just the recycling process itself to the facets of the larger system in which it fits (Fig. 1). Knowledge of the kind of recycling process that will be applied to a product (e.g., hydrometallurgical, pyrometallurgical) can affect the design choices that enhance recyclability (DfR), elaborated further below. Recycling often consists of multiple processing stages. In ideal (but uncommon) situations, DfR can enable direct reuse of disassembled, intact, recovered components (denoted by the dashed blue line in Fig. 1), which minimizes intermediate losses and overall recycling energy use.

Although multiple studies discuss DfR in some capacity, few studies have attempted to identify specific strategies for implementing it [4, 7–9]. Others provide broad DfR strategies and comment on DfR’s relationship within broader sustainability practices [2, 5]. This article synthesizes and extends identified DfR principles based on a review of published industrial and academic best practices as well as consultation with experts in the field; these experts are listed in the Acknowledgements. The resulting high-level DfR principles are meant to be useful for stakeholders—such as research and development (R&D) managers, analysts, and policymakers—in informing and promoting decisions that facilitate DfR.

We focus on the application of DfR to several clean energy technologies because of their recent and projected growth (e.g., to fulfill renewable portfolio standards, \(^1\) 100% renewable energy).

\(^1\) http://www.ncsl.org/research/energy/renewable-portfolio-standards.aspx.
renewable energy targets,\(^2\) and other initiatives\(^3\) and to help avoid future waste-management challenges. Cumulatively, by 2050, estimates project 78 million tonnes of raw materials embodied in the mass of EOL photovoltaic (PV) modules (which are \(\sim 95\%\) crystalline-silicon modules), 12 billion tonnes of wind turbine blades, and by 2030, 11 million tonnes of lithium-ion batteries [10]. Thus, we provide DfR principles specific to crystalline-silicon PV modules, batteries like those used in electric vehicles (EVs), and wind turbine blades, along with a set of broader principles applicable to all three of these technologies and potentially others. These principles are based on the current state of knowledge and technologies, and revisions will be needed as the field continues to evolve. The article concludes with a discussion of commercial DfR applications in clean energy and a summary of the principles.

Principles Applicable to All Three Selected Clean Energy Technologies

Based on our literature review—see the Supplementary Information (SI)—and consultation with experts, we identify nine DfR principles that broadly apply to crystalline-silicon PV modules, EV batteries, and wind turbine blades, as described below. Readers could consider whether these broad principles might also apply to other clean energy technologies. Principles specific to the selected clean energy technologies are provided in “PV DfR Principles” section (crystalline-silicon PV), “EV Battery DfR Principles” section (EV batteries), and “Wind Turbine Blade DfR Principles” section (wind turbine blades).

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\(^2\) https://news.energysage.com/states-with-100-renewable-targets/.

\(^3\) For example, http://there100.org/companies.
Product Requirements Such As Functionality, Longevity, Reliability, and Cost Are Critical for Market Acceptance; DfR Should Support or Enhance These Aspects but May Result in Trade-Offs Between Recyclability and Product Performance and Cost

DfR products must be accepted in the market for the recycling benefits to be achieved, which requires that the function, reliability, and cost be acceptable to consumers. In some cases, improving a product’s recyclability may also improve its performance and/or cost, but in some cases these different objectives conflict. If a high degree of recyclability impairs a product’s commercial viability, designing for less recyclability may be necessary [4]. “Crystalline-Silicon PV DfR Principles” section (Principle 4) provides an example of such a trade-off, in relation to the laminate-free NICE-PV module that sacrifices some efficiency to facilitate disassembly and recycling. Conversely, if a product is subject to recycling targets (e.g., due to government policies), lower performance or higher cost may be trade-offs for greater recyclability.

Material Choice and the Ability to Liberate Separate Materials Are Critical to DfR Outcomes

DfR outcome metrics like material recovery rate and purity of recovered materials depend on the choice and distribution of materials within the product as well as use of adhesives (see Principle 6) [4, 7, 8]. Materials entering secondary material markets must meet purity specifications. If incompatible materials are used in a product, but are not adhered together, they can sometimes be adequately separated to avoid commingling of the recycling output streams—for instance, when the incompatible materials are isolated in easily separable components. However, even if incompatible materials are in the same component, they might be separable during a recycling process. For example, shredding might be able to liberate and separate copper (Cu) and iron sufficiently from a product to provide a steel mill with a stream of recycled iron that is acceptably uncontaminated with Cu [7].

Recycling Outcomes Can Be Enhanced by Minimizing Hazardous Materials in Products or Making These Materials Completely Recoverable via DfR

Hazardous materials should be avoided when possible (e.g., when not contrary to Principle 1) so the product is not classified as hazardous waste at EOL, to reduce the cost of recycling (for handling and treating hazardous materials), and to reduce environmental impacts. When hazardous materials must be used, DfR that enables full and separately controlled recovery of these materials can avoid contamination of recycling outputs and characterization of the outputs as hazardous waste. For example, recycling rates for lead-acid car batteries exceed 90%. Most of the lead (Pb) is recovered and used in new battery applications, which keeps this hazardous material out of landfills and avoids contaminating other waste streams [11]. Similarly, cadmium telluride (CdTe) PV modules are subject to a specialized process that recovers Cd and Te, which prevents environmental and human health hazards related to Cd [12, 13].

Minimizing and Managing Hard-to-Recycle Materials Can Improve Overall Recycling Yield

Hard-to-recycle materials vary by recycling situation and may include materials degraded by recycling or those not recyclable more than a few times, such as treated wood [4, 5]. When such materials must be used, DfR that makes them easy to isolate early in the recycling process can facilitate recycling of the product’s other materials.

Minimizing Non-Reversible Adhesives or Similar Bonds, Especially Over Whole Surfaces and for Dissimilar Materials, Can Facilitate Disassembly and Material Liberation

Non-reversible adhesives or similar difficult-to-break bonds may impede product disassembly and material liberation [4]. This is especially true when two dissimilar and incompatible materials are bonded together; however, in some cases, such bonds can facilitate recycling, such as when strategically placed welds (non-reversible) provide a better path of preferential breakage and liberation during shredding, compared with using bolts (reversible).

Design for Disassembly (DfD) Can Improve Recyclability

DfD promotes modular product construction, which can facilitate separation and then recycling of individual component groups at EOL (as well as repair of components individually, as opposed to needing to disassemble a whole device to repair the faulty component). In addition, product disassembly trials by manufacturers can reveal implications of various aspects of the product’s design with regard to DfR, such as choice and location of fasteners and joining methods [9, 14], and these insights can be applied in
designing a recycling process. Design for remanufacturing exhibits similar synergies with DfR [4, 5].

**Being Able to Estimate Improvements in Recyclability and Economic and Environmental Impacts Due to DfR Is Important for Continuous Improvement, Identification, and Weighing of Trade-Offs and Communicating Value**

In recent years, methods designed to quantitatively [7–9, 14] or qualitatively [4] analyze product recyclability have emerged. These methods, which vary in complexity and ease of use, are most useful when validated with experimental observations, and they can continue to be improved. DfR can be beneficial even if product designers can only assess its impacts at an informal level [4, 7–9, 14]. Benefits from analyzing effects of recycling include the ability to learn and continuously improve, to identify trade-offs and weigh their effects for future redesign, and to enable communication about the value of DfR to internal and external stakeholders.

**Using Labels to Identify Recyclable and Non-Recyclable Materials Helps Recyclers Classify Feedstocks; Labeling Standardization Is Important for Uptake and Utilization**

Product labeling options include stickers with identifying information, symbols, embossed or engraved information, bar codes, radio-frequency identification (RFID), blockchain, material passports, or other methods. The label composition, placement, and application method can have positive or negative recycling implications, but the label's information is most important [4, 7, 8]. Clear labels help recyclers classify feedstocks and thus facilitate their entire recycling operations [2, 11]. Standardization is key to efficiency of use, degree and ease of uptake, and broad utilization. Confidential or proprietary information can be maintained in some systems, with potential trade-offs in ease and speed of identification by multiple parties. Labels that provide information beyond material composition—such as appropriate handling, repair, remanufacturing, and recycling strategies—can also be helpful.

**Designing Products to Use Recycled Materials Promotes Circular Manufacturing**

Although using recycled materials in a product does not inherently improve that product's recyclability, doing so bolsters the circularity of the manufacturing system and encourages DfR; building markets for recycled materials gives impetus to companies and R&D managers to enhance design of products for recycling, because the materials within the products will have greater value at EOL [2, 4, 5]. In turn, increasing the prevalence of DfR may lead to higher-quality recycled feedstocks and greater utilization of these resources.

**PV DfR Principles**

This section starts with principles focused on the world's dominant PV technology, crystalline-silicon PV modules (“Crystalline-Silicon PV DfR Principles” section), followed by a short discussion of DfR considerations for thin-film PV modules, which hold the remaining market share (“Discussion of Thin-Film PV” section). Various crystalline-silicon PV module recycling concepts exist [15–30], including three that have achieved pilot scale or larger: the hot-knife (thermal) glass separation process offered by NPC [18, 19], a mechanical-only process run by Veolia in France [20–23], and a mechanical-chemical process designed in the Full Recovery End Life Photovoltaic (FRELP) research project—see the SI [15, 24, 25]. The basic design of crystalline-silicon modules has not changed for decades, although manufacturers have created thousands of variations. Such variability presents a challenge to recycling systems and PV DfR. In addition, PV systems are expected to perform with high reliability for warranty periods of 2–3 decades under extreme outdoor environmental conditions. Such stringent performance requirements have incentivized a design philosophy leading to a sealed, durable, sandwich-like module structure that hinders separation and liberation of constituent materials. DfR for PV modules must consider current market (price, performance) and safety expectations as the starting point for any proposed changes.

**Crystalline-Silicon PV DfR Principles**

**Durable Identification of Module Construction and Composition Could Enable Safer and More Efficient Recycling Processes**

Identifying module composition and construction may permit higher tolerance for variable module designs that are otherwise suboptimal from a recycling perspective, because the recycling process can be designed to accommodate known variability. Known composition could also facilitate batch processing of categorized groups, enable isolation of problematic or incompatible chemical compositions, and avoid contamination of recycling products. Because modules may

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4 For example, www.everledger.io.

5 For example, http://www.epea.nl/circularity-passports/ and https://www.bamb2020.eu/topics/materials-passports/.
outlive their manufacturers, it would be helpful for labeling to be durable (on the scale of decades) and for any linked databases of construction or composition to remain accessible after a manufacturer goes out of business. Steps in this direction are being taken. Although currently limited, a new voluntary sustainability certification standard, NSF-457, includes provisions for identifying PV module compounds [30]. Emerging regulations in France surrounding minimization of PV’s carbon footprint will require manufacturers to identify module composition details [31, 32].

**Backsheet Composition Has Particularly Important Implications for Recyclability**

The design of PV module backsheets has implications for cost, performance, and recyclability [25]. One important issue relates to backsheets containing fluorinated polymers, which produce hazardous fluorine (F) gases under thermal processing and thus increase thermal recycling costs or restrict treatment options. For example, use of pyrolysis is challenged because the pyrosis oil becomes contaminated with F [25, 33–37]. Some segments of the PV industry have reduced module F content to reduce production costs (see the SI) [38]. Recently, recyclability has become a driver of F-free backsheet manufacturing [37], particularly for higher-end modules. Transparent backsheets exist in both fluorinated [39] and F-free [40] varieties. If thermal processing is not used, backsheet composition has less effect on recyclability.

If fluoropolymers must be used in backsheets for a particular module design, lower F content is preferable for three reasons. First, air emissions from thermal processing can be controlled at lower cost owing to lower use of reagent to neutralize F. Second, the resulting gases will be less corrosive. Finally, Cu smelters will pay more for recovered materials with lower levels of F contamination.

The cost implications of recycling fluorinated backsheets have been debated. Some stakeholders say the incremental cost impact is limited, particularly where regionally available thermal treatment infrastructure is already equipped for fluoropolymers. However, the incremental cost could reach 1 U.S. dollar (USD) per module when a primary Cu smelter treats an F-bearing backsheet [41], which could be roughly equivalent to the recovered Cu value per module [30], thus reducing the economic feasibility of recycling. When a module’s polymer content represents a potentially salable recycling output (as with the Veolia process), backsheet content considerations depend on the local or regional recycled plastics market [20].

Future recycling processes may be able to remove fluoropolymers from backsheets mechanically [19, 42, 43]. In such a process, designs that minimize fluoropolymer coatings on the inner layers of the module would likely be easier to treat. If a future recycling option can remove and isolate a thin (~50 μm) layer from the exterior of a backsheet (of fluoro-polymer and/or non-fluorinated base layer materials), this could improve the tolerance of recycling systems to the use of exterior fluoropolymer layers in future module designs [44]. The SI contains descriptions of R&D aimed at mitigating recyclability problems due to fluorinated backsheets. Glass/glass or double-glass PV modules that eliminate the backsheet in favor of a second layer of glass would likely bypass the F issue.

Beyond F content, other backsheet attributes can affect module recyclability. For example, polymers containing nitrogen can result in nitrogen oxides during thermal treatment. Nano-titania m dioxide (TiO2) pigment may become subject to additional regulatory hurdles, particularly in thermal treatment applications [44]. The recycling implications of using carbon black pigments or conductive backsheets are unclear [45].

**Metal Choices Can Have Significant Impacts on Recycling Processes and Costs**

Substitution of silver (Ag) metallization with Cu/nickel (Ni) presents recycling trade-offs, which depend in part on the recycling process used. Ag, one of the largest sources of module recycling revenues, is also one of the most resource-constrained PV feedstocks. Reducing Ag use in modules may be necessary to sustain rapid global PV growth [46–50]. Partial substitution with Cu/Ni is less favorable than total substitution in recycling processes that use leaching, such as FRELP, owing to the additional complexity of handling both Ag and Ni in the leach solution. The presence of Ag makes the mixture unsuitable to alternative and less expensive leaching chemistries [51]. Costs for recycling processes without leaching, such as the hot-knife/Cu smelter process, likely would be less affected by partial or total substitution of Ag with Cu/Ni. In some cases, Ni can be recovered when present in a Cu smelter feed, and penalties likely would not be incurred if Ni concentrations are below 1% mass in the dry smelter feed [52].

Various candidates could replace Pb in module solder alloys, with a range of impacts on recyclability. Cu or Ag could increase recycling revenues, although using Ag may strain feedstock availability and increases cost. Tin (Sn) is effectively interchangeable with Pb in terms of recycling impacts; Sn is scarcer and more expensive than Pb, but less so than Ag. Bismuth (Bi) presents the most recycling problems among Pb-replacement candidates. Bi can result in higher penalties when selling Cu-bearing recycling outputs to a Cu smelter, reaching 3–10 USD per tonne of smel ter feed per 0.1% of excess Bi for a typical feed consisting of 22%–30% Cu [52]. In contrast, Sn and Pb are tolerated at much higher concentrations (3%–5% Pb + Sn + zinc) and
incure lower penalties (1 USD per tonne per 1% mass) [52].
Bi interferes with downstream electro-refining of smelter
Cu outputs, increasing complexity and cost [52]. Failure
to remove Bi results in lower electro-refining efficiencies,
higher energy intensity for PV recycling, and potential con-
tamination of the Cu cathode; the mechanical properties of
Cu products suffer with even 0.2% mass Bi contamination
[53]. Cu refiners rely on a sulfuric acid process, but the same
issues may manifest in more sophisticated PV recycling
flowsheets, such as FRELP’s nitric acid-based process,
and could require further recycling stages. In addition, not all
Bi-isolation methods yield salable Bi outputs, so circularity
of Bi is not assured. Pb and Sn interfere much less with such
circuits and can be readily managed in FRELP’s nitric acid-
based process [15]. Should Bi be required in crystalline-
silicon PV applications [54], some portion of the recycling
impacts could eventually become more tolerable [55]. See
the SI for additional discussion of this issue.

Minimizing Laminate Use or Using Reversible
Laminates Can Facilitate Disassembly of PV Modules

Delamination poses a challenge to many PV recycling pro-
cesses. The FRELP and hot-knife processes use high tem-
peratures to volatilize the laminate from the cells [15, 18,
24, 25]. The Veolia process separates materials laminated
together through a complex mechanical process. This pro-
cess may have more tolerance for existing laminates but
would likely perform better without them owing to reduced
cross-contamination of recycling products, improved libe-
ration, and more homogenous plastic outputs [20–23].

The manufacturers of the NICE-PV module offer a
laminate-free design that facilitates disassembly and recy-
cling, but module efficiency declines because of the air gap
and electrical connections held together by internal mod-
ule vacuum pressure; changes in temperature and altitude
would change the electrical efficiency [56–60]. One study
demonstrated that the NICE design also enabled relatively
easy recovery of intact wafers and glass panes from a simu-
lated EOL module, which is currently infeasible for modules
with laminates [60]. TPedge is another laminate-free module
design. It uses adhesive pins, which cover only 0.02% of
the cell surface, and soldered electrical connections that require
no specific under-pressure but might exhibit some sensitivity
to temperature or altitude. Both NICE and TPedge eliminate
laminate from the face of the cells with the help of edge-
sealing methods like silicone. Still, some glue-like material
is required on portions of the front glass and rear materials.
Although recyclability has not been specifically demon-
strated on a TPedge module, the design indicates it could be
more successful than most typical modules [61–63].

Ideally, future laminate designs would offer the necessary
durability but also be more easily reversed during recycling
than the currently typical module laminate, ethylene vinyl
acetate (EVA) [64]. Other materials, such as silicone,
may offer benefits accompanied by trade-offs in the form
of increased unit cost or different life cycle impacts [65].
A study found, for instance, that silicone laminates could
increase short-circuit current density and cell efficiency
under certain conditions [65, 66]. This result may, how-
ever, not apply to dusty conditions [66]. Moreover, silicone
encapsulants have been found to lose adhesive properties in
extreme weathering conditions [67].

Decreasing the Number and Complexity
of Module Materials Presents Trade-Offs Related
to Recyclability and Economics

Two trends in PV module designs exemplify trade-offs with
regard to reducing the number and complexity of materials.
Frameless modules are one trend. PV modules are typically
designed with frames, but they can be designed without
frames. Framing helps protect the module during transporta-
tion, installation, and EOL removal while easing the installa-
tion process and providing torsional rigidity throughout the
life cycle. Frameless modules are more prone to breakage,
although certain transportation strategies and, for instance,
reusable corner protectors can reduce breakage. On the other
hand, frameless modules simplify recycling. De-framing a
module adds a recycling step and increases the potential for
glass and cell breakage [68]; however, the frames are rela-
tively easy to recover, and the aluminum (Al) can add more
than 2 USD/module in recycling revenue [16, 30]. Thus,
the frame’s impact on recycling economics is comparable to
module Ag content (~2.70 USD/module [30]), particularly
for recycling processes such as FRELP and hot-knife [15,
18, 24, 25]. As yet another trade-off, frameless modules are
lighter and more compact and can cost less to transport at
EOL. Removing the frame (whether in design as frameless
or through frame removal at EOL) can increase up to sev-
enfold the number of modules that fit in the same container
[20].

Glass/backsheet versus glass/glass module designs pre-
sent other trade-offs. Glass/glass designs increase the poten-
tial glass cullet revenue per module and eliminate use of a
backsheets, which is often fluorinated [16, 30]. If the rear
and front glass are of different grades and are not isolated from
each other throughout the recycling process, introducing the
multiple grades into outputs may degrade the glass recy-
cling economics. Experimental recycling data are sparser
for glass/glass modules than for the typical glass/backsheets
varieties. Because the mechanical properties of glass differ
from those of polymer backsheets, recycling processes that
intentionally or coincidentally exploit such attributes could
be impacted in unpredictable ways. The mechanical Veolia
process has undergone preliminary testing with glass/glass
modules, and future glass/glass recycling is possible with some retrofittings, according to a La Mia Energia representative (personal communication, 2019).

Discussion of Thin-Film PV

The crystalline-silicon PV DfR principles pertaining to structural aspects of a module generally remain valid for a range of thin-film module chemistries; however, the metal-recovery stages of the recycling process may vary as a function of underlying module chemistry. A copper-indium-gallium-selenide (CIGS) module may require different treatment [69, 70] than CdTe [12, 13] or perovskite [71–73] modules. First Solar manufactures and also recycles their CdTe modules [12]. Integrating these functions in the same facility and supply chain enabled the establishment of a best practice for DfR implementation: First Solar’s recycling division is part of the approval chain for new module designs, ensuring recyclability is considered in developing new module designs. Additionally, raw material values are higher for CdTe than for crystalline silicon, inducing interest in enhancing the security of their supply chain by participating directly through EOL recovery of precious metals like Te. Handling of Cd, a metal with high toxicity, must be carefully controlled to ensure no environmental release, something that the company had to prove early on to ensure their social license to operate. Laboratory-scale recycling of several CIGS and perovskite module designs has been demonstrated [69–73]. Despite their infancy, circularity appears possible for perovskite PV, despite concerns about Pb content [71, 72].

EV Battery DfR Principles

Many battery DfR principles depend on the battery chemistry and recycling flowsheet [11, 36, 74–77]. This section lists several principles applicable to current and next-generation EV batteries. These principles focus on the battery exterior housing, because they can be applied well to the range of different battery types. Overall, standardization of product size and design can help streamline and optimize operations for battery recyclers, but it is not the current reality. Internal configuration and material choices are also important recyclability considerations, but the variability of these characteristics hinders identification of broadly applicable principles. The principles listed here are generally applicable to batteries sized for EVs and smaller batteries. Where possible, commentary on lithium-ion batteries is provided, but these batteries consist of a highly diverse set of chemistries and recycling options [75–77]. The principles are generally less applicable to large-scale vanadium redox flow batteries (VRFBs); performance, durability, and recyclability are complementary, rather than competing, criteria in this application—at least with regard to the vanadium (V) electrolytes. In addition, the larger the scale, the better the capacity for built-in equipment redundancy, isolation, and repairability during operation, all of which further improve recyclability [78, 79].

Recyclability Is Usually Improved When Removal of the Battery Is Made Easier

Easy battery removal improves recyclability of the battery (and the host product), which increases the probability that batteries will end up at dedicated recycling outlets [4, 9, 11, 14, 80–82]. However, in some situations (e.g., with EVs), a hard-to-remove battery may enhance recyclability by making it more likely that only trained personnel attempt removal, which may improve safety and control over the recycling supply chain.

Clear Identification of Battery Chemistries Can Mitigate Recycling Safety Hazards

Improper mixing of battery chemistries during recycling can be a safety hazard. Conversely, avoiding such mixing can improve the probability of any given battery material being reintroduced into a manufacturer’s supply chain. Clear identification of chemistries can mitigate this issue, using approaches such as color coding, RFID, barcodes, and ultraviolet or infrared scanning. Recent advances in sensor-based sorting technology may help battery recyclers use such identifiers to reject incompatible batteries early and cost-effectively [11]. The battery industry has historically resisted color coding owing to concerns about loss of brand recognition, but even partial adoption of such an approach (e.g., on battery areas that are never seen by consumers) could offer recycling benefits.

Matching the Battery Exterior’s Chemical Compatibility to the Intended Recycling Process Can Avoid Unwanted Chemical Interactions and Facilitate Recycling

It is important to ensure chemical compatibility between the exterior battery materials and the recycling process, because some exterior materials may not react well to some chemical or high-temperature processes. In some cases, using a chemically ideal exterior may enable skipping of early physical separation. Alternatively, if the material can be physically isolated upstream of the chemistry-intensive recycling stages, chemical incompatibility may not be relevant; for example, removing an Al exterior casing will limit the amount of Al exposed to chemical processing.
When durable, rigid, non-polymer, non-detachable exteriors are required, some materials are more amenable to existing recycling systems. It can be advantageous to isolate exterior or internal materials prior to downstream recycling in an attempt to separate at the earliest steps and provide the best opportunity for homogeneous recovered materials. For example, when subjecting lithium (Li), Ni, or cobalt (Co) compounds to acid leaching, it is beneficial to separation and performance outcomes to remove any Al exterior material before the acid-leaching process [36], to limit the amount of Al exposed to leaching.

Rigid Polymer Exteriors Are Often Well-Suited to Recyclability

Rigid, durable metal exteriors, especially those that cannot be readily detached, can hinder access to internal components and damage recycling equipment, such as crushers. Rigid polymer exteriors, by contrast, pose little threat to recycling equipment and can offer high material recovery; they are typically used for lead-acid battery exteriors. Some of these polymers float, which can facilitate dedicated polymer separation and recovery [11]. Fiberglass, another exterior material choice, currently has poor recyclability [80, 81]. Flexible or foil-like exteriors likely have fewer negative implications for recyclability, but this cannot be assumed for all situations. Metal exteriors are typical for EVs owing to structural strength requirements. Battery designs that can facilitate removal of the exterior (e.g., bolts instead of welds) can facilitate the recycling process.

For exteriors subjected to crushing or shredding, magnetic materials can be rejected pre- or post-crushing, or eddy current separators can enable rejection of Al/magnesium (Mg) alloy exteriors [83]. Magnetic or eddy current separation could also be used to remove higher-cost Ni/Co alloys [83], which—with additional processing—likely would be chemically compatible with many lithium-ion and some nickel-metal-hydride recycling flowsheets. Carbon fiber exteriors may be acceptable when thermal treatment is considered [84], if adding carbon does not hinder the intended chemical reactions. If mechanical separation of the exterior is intended, then carbon fiber is less desirable owing to the current challenges of recycling broken carbon fiber into new applications. Finally, titanium and austenitic stainless steels do not respond to magnetic or eddy current separation [83] and thus lack suitable physical rejection methods, making them more likely to increase crushing equipment wear; methods would need to be identified to ensure these materials are not sent into the shredder.

Using F-free Binders or Otherwise Minimizing F Content Can Facilitate Recycling

The F in fluoropolymer binders may hinder recycling, but few F-free binder substitution candidates exist. However, if other sources of F are already present in lithium-ion batteries (e.g., lithium hexafluorophosphate [LiPF₆] electrolyte), substitution with a lower F-content (rather than F-free) fluoropolymer may be easier to rationalize, because F removal will still need to be included in EOL material-recovery processes.

Reducing the challenges of separating F from valuable materials at EOL can improve the economics of material recovery. In nickel-manganese-cobalt (NMC) lithium-ion batteries, these metals and Li are present in the “black mass fraction,” which is attached to electrode foils (such as Al) via a binder. The black mass fraction represents most of the battery’s rare metals and much of the Li content, and thus much of its value [36]. For this case, recovery of the black mass fraction requires separation from the Al electrode foils as well as the F binder; however, the prospects for maximum recovery decrease with increases in thermal treatment temperatures. There are some recycling advantages to enabling lower-temperature and faster binder destruction via substituting a polyvinylidene difluoride (PVDF) binder with a more readily volatilized fluoropolymer. Additionally, it is important to consider what types of materials might be volatilized in these thermal reactions to minimize hazardous emissions.

When considering alternatives to PVDF-based binders, materials based on terpolymer of tetrafluoroethylene, hexafluoropropylene, and vinylidene fluoride (THV) should be avoided. THV has a higher F content than other binders and is not prone to significant thermal decomposition until approximately 700 °C [85]. This forces the use of unfavorably high temperatures for thermal treatments for NMC lithium-ion batteries outside of recycling options that include smelters.

Wind Turbine Blade DfR Principles

Current wind turbine recycling challenges mostly focus on the blades and, to a lesser extent, the magnets, because the remaining turbine components have simpler designs and are made of more homogenous materials, in particular easy-to-recycle metals. The following DfR principles focus on blades. The SI discusses the potential DfR implications of magnet innovations. Ongoing developments in materials science, manufacturing, and recycling processes may radically shift the particulars of any wind DfR principles, and thus the principles should be reviewed regularly to ensure applicability.


**Some Blade Materials Are More Recyclable Than Others, but Blade Technologies Are Changing**

Composite materials, such as fiberglass and carbon fiber, are currently less recyclable than homogenous materials, because they are not easily separated into homogenous materials and at best are ground up and used as either filler or in other down-cycled products (Phuon Anh Vo Dong, 2018). Research is underway to evaluate blade compositions, fabrication methods, and recycling processes.9

Because recycling of wind turbine blades is currently challenging and not commonly done, reuse of blades is another circular economy option to consider. The current large and ever-increasing size of the blades, however, likely makes this option impractical. The large size of the blades also hinders recycling, but on-site EOL size reduction could make transport of blades to recycling facilities easier [86].

**Minimize Use of Additional Blade Materials Other Than Resin and Fibers**

Research is underway6 to evaluate the potential for recovering composite resins using thermoplastics in place of thermoset materials; further progress could provide a favorable recycling option for wind turbine blades. Short of existing material-recovery options, DfR would suggest prioritizing materials with the highest potential for energy recovery (while considering fire safety) though this priority is only in relation to disposal since energy recovery is the least favored CE pathway (see Fig. 1). Carbon fiber materials consist nearly entirely of reactive combustible material, whereas traditional fiberglass leaves behind up to 40% mass in an ash fraction. While combustible, fiberglass requires additional fuel to combust, and thus is not desirable as a fuel substitute except in cement operations where the ash fraction is a cement filler [87, 88]. However, carbon fiber production has a very high energy intensity, so the life cycle perspective should be considered to ensure an overall benefit.

**Some Blade Designs Enable Retention of Material Value After Recycling**

Designing wind turbine blades such that materials maintain value after recycling is a key research need. For example, recycling processes that can recover long, straight fibers may provide higher value and performance potential, compared with processes that can only recover odd geometries which have limited uses for other designs or applications.

**Commercial Implications of DfR**

The successful application of DfR principles offers potential for environmental benefits as well as economic benefits to manufacturers and end users. Emerging business models such as product-as-a-service (PaaS) may enable manufacturers to leverage recyclability to reduce prices beyond what competitors can offer with comparable but less-recyclable products.

Rentable V electrolytes for VRFBs are one prominent clean-energy example of a PaaS business model [89]. Because the vertically integrated V miner/electrolyte manufacturer knows that the V will remain recoverable and recyclable after 10–20 years of operation, they are willing to enter into rental contracts with an end user of the battery [90]. If the V miner/electrolyte manufacturer rents V to the end user, both the battery maker and end user are decoupled from full exposure to the volatile cost of V. This reduces the end user’s upfront VRFB purchase cost (by as much as 30% in one estimate [90]), making it competitive with the cost of lithium-ion batteries [90]. The result is a more circular economy with regard to material flow and a business model in which a finite mineral reserve translates into recurring annual payments for the V miner long after digging has ceased. Such alternative business models inherently rely on assurance that EOL collection and recycling are possible.

To maintain a high degree of recyclability, product developers must maintain a working awareness of potentially applicable recycling processes. Thus, DfR is easier to implement when a manufacturer recycles its own products, whether through PaaS models or through extended producer responsibility schemes like those for lead-acid batteries in the United States or PV under Europe’s Waste Electronic and Electrical Equipment (WEEE) Directive—or, even more directly, First Solar’s takeback and recycling of its modules.

**Conclusion**

Table 1 summarizes the DfR principles described in this article. Because DfR is product and situation specific, these principles must be adapted to particular cases. For example, DfR may need to consider the recycling capabilities of regions in which a product is likely to reach EOL [9]. Specific technical considerations are also important, such as whether the product will require only mechanical separation or whether it will require thermal processing.

More generally, realizing a circular economy requires attention to many factors and coordination among numerous stakeholders. Recycling and DfR are important aspects of this effort, but they should be considered after other circular approaches—such as design for reuse, product longevity,

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6 https://iacmi.org/technology-areas/wind-turbines-application-center/
Table 1 Summary of DfR principles

Principles applicable to all three clean energy technologies
1. Product requirements such as functionality, longevity, reliability, and cost are critical for market acceptance; DfR should support or enhance these aspects but may result in trade-offs between recyclability and product performance and cost
2. Material choice and the ability to liberate separate materials are critical to DfR outcomes
3. Recycling outcomes can be enhanced by minimizing hazardous materials in products or making these materials completely recoverable via DfR
4. Minimizing and managing hard-to-recycle materials can improve overall recycling yield
5. Minimizing non-reversible adhesives or similar bonds, especially over whole surfaces and for dissimilar materials, can facilitate disassembly and material liberation
6. Design for disassembly (DfD) can improve recyclability
7. Being able to estimate improvements in recyclability and economic and environmental impacts due to DfR is important for continuous improvement, identification, and weighing of trade-offs and communicating value
8. Using labels to identify recyclable and non-recyclable materials helps recyclers classify feedstocks; labeling standardization is important for uptake and utilization
9. Designing products to use recycled materials promotes circular manufacturing

Crystalline-silicon PV principles
1. Durable identification of module construction and composition could enable safer and more efficient recycling processes
2. Backsheet composition has particularly important implications for recyclability
3. Metal choices can have significant impacts on recycling processes and costs
4. Minimizing laminate use or using reversible laminates can facilitate disassembly of PV modules
5. Decreasing the number and complexity of module materials presents trade-offs related to recyclability and economics

EV battery principles
1. Recyclability is usually improved when removal of the battery is made easier
2. Clear identification of battery chemistries can mitigate recycling safety hazards
3. Matching the battery exterior’s chemical compatibility to the intended recycling process can avoid unwanted chemical interactions and facilitate recycling
4. Rigid polymer exteriors are often well-suited to recyclability
5. Using F-free binders or otherwise minimizing F content can facilitate recycling

Wind turbine blade principles
1. Some blade materials are more recyclable than others, but blade technologies are changing
2. Minimize use of additional blade materials other than resin and fibers
3. Some blade designs enable retention of material value after recycling

and remanufacturing—that provide better material value retention (shown as tighter circles in Fig. 1). DfR is inherently challenged by the time that elapses between manufacturing of a product and the product’s EOL. A conservative approach is to design products with some assurance of the recycling system’s ability to handle the materials at EOL. Manufacturers can take calculated risks based on knowledge of potential future recycling methods, and they can mitigate risk further by becoming directly involved in planning future recycling capabilities or even operating their own recycling operations for their products (such as with the First Solar example above). The market for materials and the supply and demand will drive the success of recycling. Economics, product performance, and environmental impacts must be considered; recycling at EOL may not always be the best solution for the existing manufacturing and recycling ecosystem. Ultimately, DfR is desirable to achieve system-wide goals, such as ensuring material supply, maximizing material value, or minimizing life cycle impacts. There is a need for analytical tools to better evaluate the impacts of recycling, DfR, and other circular-economy strategies.

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Compliance With Ethical Standards

Conflict of interest  On behalf of all authors, the corresponding author states that there is no conflict of interest.

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