Biodegradable, flexible silicon nanomembrane-based NO\textsubscript{x} gas sensor system with record-high performance for transient environmental monitors and medical implants

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Abstract

A novel transient electronics technology that is capable of completely dissolving or decomposing in certain conditions after a period of operation offers unprecedented opportunities for medical implants, environmental sensors, and other applications. Here, we describe a biodegradable, flexible silicon-based electronic system that detects NO species with a record-breaking sensitivity of 136 Rs (5 ppm, NO\textsubscript{2}) and 100-fold selectivity for NO species over other substances with a fast response (~30 s) and recovery (~60 s). The exceptional features primarily depend on not only materials, dimensions, and design layouts but also temperatures and electrical operations. Large-scale sensor arrays in a mechanically pliable configuration exhibit negligible deterioration in performance under various modes of applied loads, consistent with mechanics modeling. In vitro evaluations demonstrate the capability and stability of integrated NO\textsubscript{x} devices in severe wet environments for biomedical applications.

Introduction

Electronic materials and devices that are biologically benign or mechanically elastic have been utilized in versatile applications, including electronic skin (E-skin) and biomedicine, for the diagnosis and treatment of diseases due to their advantages of nontoxicity, bioresorption, and mechanical properties similar to those of human skin and organs. Published examples involve sweat-based analysis\textsuperscript{1–3}, epidermal virtual reality\textsuperscript{4,5}, self-healing electronics\textsuperscript{6,7}, optoelectronic elements\textsuperscript{8,9}, and neuromodulation\textsuperscript{10–13}. Although the technological advances in the corresponding areas delivered innovative output products, a relatively unexplored research opportunity exists in detectable species indicative of occurrences in organisms that may affect the prevalence and severity of human health, such as infections, injuries, diseases, or environmental exposures. One of the key signaling elements is nitric oxide (NO), which is produced naturally in the human body and plays an essential role in many aspects of maintaining health by relaxing or widening blood vessels to enhance blood flow, allowing nutrients and oxygen to travel to every part of the human body\textsuperscript{14,15}. In non-biological cases, highly reactive nitric oxide can be transformed into nitrogen dioxide (NO\textsubscript{2}) with other participants, contributing to acid rain deposition and harmful ozone generation\textsuperscript{16,17}. Research approaches for NO species in previous works have explored high-performance channel materials such as graphene, carbon nanotubes, polymers, and metal oxides\textsuperscript{18–20} or modifications to these structures for a high surface-to-volume ratio\textsuperscript{21–25}, while several issues, including rigidity, low sensitivity, poor
selectivity, high power consumption, and nonbiodegradability, still need to be considered for wearable and implantable electronic systems. Here, we report a flexible and bioresorbable single-crystal silicon nanomembrane (SC-Si NM)-based NOx sensing system operating at room temperature with exceptional sensitivity and selectivity. The results focus on detailed studies of electrical responses of Si NMs under various conditions, and of the mechanical properties of the electronic system with theoretical considerations. In vitro assessments that incorporate semipermeable membranes provide information on their potential applicability in disposable environmental monitors and bioresorbable medical implants.

**Experimental procedure**

**Fabrication of a transient NOx (NO/NO2)-sensing system on various substrates**

Device fabrication began with thinning down SC-Si NMs on silicon-on-insulator (SOI, SOITEC, France) wafers (top silicon ~300 nm in thickness). Repeated oxidation of SC-Si NMs at an elevated temperature (1100 °C) and removal of the thermally grown SiO2 by wet etching with hydrofluoric acid (HF, 49%, J. T Baker, USA), allowed to reach a desired target thickness of Si NMs (~100 nm). Phosphorus doping at 950 °C with spin-on dopant (SOD, Filmtronics, USA) formed highly doped regions for electrical contacts of an NOx sensing system. Undercut wet etching of oxide released the thinned SC-Si NMs from the carrier wafers, enabling transfer printing onto a PMMA/diluted PI-coated temporary substrate. Active areas of the SC-Si NMs for NOx gas, humidity, and temperature sensors were defined by sulfur hexafluoride (SF6)-based reactive ion etching, followed by deposition of interdigitated electrodes (IDEs) with a layer of magnesium (Mg, ~300 nm) using an electron-beam (e-beam) evaporator (VER5004, SNTEK, Korea). Formations of patterned dielectric layers and contact pads were generated by wet etching using buffered oxide etchant (BOE, 6:1, J. T Baker, USA) and e-beam evaporation. A geometry of patterned bridges was employed to facilitate lifting off the device, enabling the release of the system from the temporary substrate. After removal of the bottom PI layer, the device was transfer printed onto degradable substrates for a complete transient electronic system.

**Characterization of soft, biodegradable NOx sensors**

The electrochemical properties of a silicon-based gas detector were measured using an electrometer (Keithley 2636B) under a DC bias voltage of 0.01 V. Experiments were conducted in a temperature-adjustable chamber (RT ~50 °C, Lindberg, USA) with a constant flow rate of 1000 sccm controlled by mass flow controllers (MFCs). Electrical responses of sensors were determined by measurements of baseline resistances in dry N2/air and saturated resistances after exposure to target gases for 500 s. The responses to reducing and oxidizing gases were defined as R0/Rgas and Rgas/R0, where Rgas and R0 denote the resistances of the target gases and dry N2/air, respectively.

**Cell culture and M1 activation**

RAW 264.7 cells were cultured in Dulbecco’s modified Eagle’s medium (DMEM) supplemented with 10% fetal bovine serum (FBS), 100 U/ml penicillin, and 100 μg/ml streptomycin. Cells were incubated at 37 °C with 5% CO2 in a humidified incubator. To induce M1 activation, 100 ng/ml LPS (lipopolysaccharide, Sigma) and 20 ng/ml IFN-γ (interferon-gamma, Peprotech) were added to the culture medium, and it was incubated for 24 h.

**DAF-FM assay for NO measurement**

After M1 activation, cells were incubated in L-arginine-free DMEM (Thermo, 88364) for 30 min, collected using a cell scraper and replated in 48-well plates. To measure NO production in cells, 10 μM DAF-FM DA (Sigma, D1946) was added to medium with or without 150 μM L-arginine, and the system was incubated for 30 min. The fluorescence intensity of DAF-FM-stained cells was measured using a Cytation 3 Cell Imaging Multi-Mode Reader (Bio-Tek).

**Immunofluorescence microscopy**

Cells were fixed using 4% paraformaldehyde for 10 min at 4 °C and permeabilized with 0.05% Triton X-100 for 5 min. After blocking with FBS (10%, v/v)-supplemented PBS for 30 min, cells were incubated with anti-iNOS antibody (1:500, Abcam, ab178945) for 60 min. Cells were then incubated with secondary antibodies (1:500, Bethyl, A120-100D4) for 60 min in the dark. The nucleus and F-actin were stained with DAPI (Invitrogen) and Alexa Fluor phalloidin (1:50, Invitrogen, A12379), respectively. The fluorescence confocal images were captured using confocal laser microscopy (A1R, Nikon) through a ×60 oil immersion objective (N.A. 1.4).

**Detection of NOx produced in M1-activated cells**

M1-activated cells were transferred into a sterile cell-culture flask containing 2 ml L-arginine-free DMEM. NO sensors were placed into the flask, and L-arginine was provided through needles embedded in stoppers. The NO sensors were connected with ACF cables to an electrometer, and all measurements were recorded through LabVIEW with the GPIB interface.

**Results and discussion**

**Integrated bioresorbable, flexible NOx (NO/NO2) sensing system**

Figure 1a illustrates examples of various adverse environmental and biological effects that may be caused by
nitrogen oxides (NO$_2$). NO has been well recognized as a biomarker for nervous, vascular, and respiratory systems$^{26-28}$, and a major air pollutant giving rise to acid rain and ozone generation after transformation into NO$_2$ as a reaction intermediate. Figure 1b provides an exploded view drawing of a soft, bioresorbable NO$_x$ sensing system that includes highly sensitive and selective silicon nanomembrane (Si NM)-based gas sensors at room temperature, and a temperature and hydration monitor to adjust the properties of the gas sensors affected by those parameters. The electronic system consists of patterned single-crystal silicon nanomembranes (SC-Si NMs, thickness ~100 nm) doped with phosphorous using spin-on dopant (SOD, Filmtronics) for reactive materials, magnesium (Mg, thickness ~300 nm) as electrodes, silicon dioxide (SiO$_2$, thickness ~100 nm) as interlayer dielectric, and an elastomeric polymer (PDMS, thickness ~20 μm) as a semipermeable membrane to maintain a stable gas sensing performance in humid and wet conditions. Sample preparation involves thinning of SC-Si NMs to a targeted value (~100 nm), doping with phosphorous at selected regions on silicon-on-insulator (SOI) wafers and transfer printing doped SC-Si NMs to a temporary foreign substrate. Electrodes and insulators were evaporated and
patterned through deposition, photolithography and etching, and transfer printing of the resulting components to a biodegradable substrate completed the whole process. In the case of measurements in humid or wet conditions, facile spin casting of PDMS was applied to the obtained sample to form a gas-permeable membrane. The biodegradable, flexible NOx sensing system laminated on a wrist appears in Fig. 1c, with a magnified image of the individual gas, temperature, and humidity sensors in the inset (more images in Fig. S1). Figure 1d shows a set of images of the temporal degradation behavior of a representative NOx sensing system assembled on a poly(lactic-co-glycolic acid) (PLGA) substrate during half-immersion in a PBS (pH 7.4) at 37 °C for 0, 1, 4, and 8 h. All constituent materials dissolved via hydrolysis to form non-toxic end products. Mg electrodes, with a relatively rapid dissolution rate, disappeared faster than other components via the reaction Mg + 2H2O → Mg(OH)2 + H2, and SC-Si NMs, SiO2, and PLGA gradually dissolved at rates similar to previously reported values while undergoing the reactions Si+ 4H2O ↔ Si(OH)4 + 2H2, SiO2+ 2H2O → Si(OH)4, and poly(lactic-co-glycolic acid) → lactic acid + glycolic acid + H2. Additional dissolution images at different stages appear in Fig. S2.

High-performance biodegradable NOx sensing system based on silicon nanomembranes

Figure 2a illustrates the fundamental mechanism of SC-Si NMs reacting with nitrogen oxides. Changes in the electrical properties of n-type semiconductor sensors resulted from variations in a depletion layer on the surface through adsorption and reaction with the oxidizing gas molecules, i.e., electrons were extracted from the SC-Si NMs as the reaction NOx (gas) + e− → NOx (ads) occurred. Figure 2b represents a measured response to 5 ppm of NO2 at room temperature (RT) with a superb sensitivity of 136 Rs (13,600%) as well as fast response (30 s) and recovery (60 s) rates, which surpass those of previously reported room-temperature sensors based on various materials, including carbon nanotubes (1.14–1.2 Rs), graphene (1.12–1.6 Rs), polymers (5 Rs) and metal oxides (2.2 Rs), and a commercial gas sensor (7 Rs, MICS-2714, Amphenol SGX Sensortech, Switzerland) (Fig. S3). Investigations at a wide range of gas concentrations indicated that the detection had linear electrical responses, and the inset (Fig. 2c) provides a detection limit (DL) of ~20 ppb, which was estimated by a linear least squares fit for data collected at room temperature. The excellent sensitivity sufficiently detects concentrations required in potential applications, such as concentrations in exhalation by asthma and halitosis patients (>100 ppb) and the atmosphere (0.6–5 ppm). The rationale that crystallinity, thickness, device layout, and electrical operation would affect the exceptional performance of the biodegradable silicon devices motivated various types of examinations. Figure 2d shows the measured responses using different kinds of silicon, such as SC-Si NMs (black), polycrystalline silicon (red, p-Si), and amorphous silicon (blue, a-Si), with identical dimensions (1.45 mm × 1.2 mm) and thicknesses (~100 nm), providing evidence of a dependence on Si crystallinity (Fig. S4). Analyses of other factors, e.g., film thickness, electrode layout, and applied voltage, are shown in Fig. S5, and dissolvable metals for coplanar interdigitated electrodes (IDEs) did not affect the functional properties (Fig. S6). The selective response to specific target gases is also an important criterion. We performed selectivity assays for well-known gases: oxidizing gases such as NO2 (black) and NO (red) and reducing gases such as NH3 (blue), CH3COOH (green, acetone), CH3OH (magenta, ethanol), CO2 (cyan), CO (navy), and H2S (purple). The resulting properties in Fig. 2e indicate that the responsiveness to NO and NO2 was at least 100-fold greater than that to other substances (detailed data in Fig. S7), which was achieved without additional physical and chemical treatments. We note that the variation in the responses between NO and NO2 gases was negligible (see Fig. S8). The temperature dependence of the sensitivity shown in Fig. 2f was evaluated to consider potential utility as medical implants and environmental monitors (black, RT; red, 37 °C; blue, 50 °C). The results were opposite to the expected behavior of conventional gas sensors, assuming that the electrical resistance/conductance of the silicon membranes would somewhat deviate from the optimum value at increased temperatures. Integration with temperature and humidity sensors can adjust the output values of gas detectors susceptible to ambient conditions for system-level functionality. The electrical characteristics of the resistive device showed a sensitivity of ~9.6 nS/°C and a temperature coefficient of ~0.006 /°C (Fig. 2g), and changes in the capacitance of the humidity component were well resolved over a wide range of relative humidities (Fig. 2h).

Mechanical stability of large-scale, flexible, and stretchable gas sensor arrays

As introduced in previous reports on deformable electronics, structures that involved device islands connected by serpentine bridges allowed devices to accommodate mechanical loads without deterioration in electrical performance. Figure 3a represents a large-scale, biodegradable sensor array (5 × 5, 1.7 cm × 1.9 cm) deployed in a mechanically soft geometry, consisting of patterned SC-Si NMs as active components, Mg conductors for contact pads and interconnects, patterned SiO2 layers for inter-layer dielectrics, and polycaprolactone (PCL) as substrate/encapsulant. The inset shows a magnified optical micrograph of an individual unit. Figure 3b, c show a set of
images under mechanical tests using several bending radii (left; flat; middle, 5 mm; right, 3 mm) and uniaxial tensile strains up to 40%, respectively. An experimental setup combined with a stretching system is shown in Fig. S9. The resulting values under those deformed conditions remained within effective ranges of 90% (Fig. 3d, bending) and 95% (Fig. 3e, stretching). In addition, the mechanical robustness and durability of the electronic system were investigated under cyclic bending with a radius of 5 mm (black) and stretching with 35% (red), as shown in Fig. 3f, and the decreases in the relative response rate in both cases after 1000 cycles were negligible. Three-dimensional finite element analysis was implemented in the commercial ABAQUS package to study the mechanical deformation of the biodegradable Si-based gas sensor upon bending and stretching. Without losing generality, a 2 × 2 array of gas sensors was considered in the simulation, and the substrate (100 µm) and encapsulation (50 µm) layers were modeled by an 8-node 3D element (C3D8R). PCL (Young's modulus, 0.25 GPa; Poisson ratio, 0.44) and semipermeable membranes (PDMS, 2.61 MPa; 0.49) were explored for the substrate and encapsulation in the bending test (Fig. 3d), whereas PDMS was used for the substrate in the tensile test (Fig. 3e). Because of their thinness, gas-sensitive Si nanomembranes (188 GPa; 0.28), Mg electrodes and interconnects (45 GPa; 0.29), and SiO2 interlayer dielectrics (66.3 GPa; 0.15) could be modeled as skin layers, as shown in Fig. S10. For bending
with a radius of curvature of 3 mm in Fig. 3g, the maximum principal strains in the layers of Mg, SiO$_2$, and Si were obtained at 1.66, 1.81, and 0.24% (Fig. S11), far below their corresponding fracture strains, which explains the mechanical stability. Upon stretching, the maximum principal strain in the skin layer occurred in the serpentine interconnect near the Si nanomembrane (Fig. 3h), and detailed strain distributions for each material are shown in Fig. S11. When the total stretchability of the system is defined at a level, where the maximum principal strain of half the width of one section exceeded the fracture strain of the metal interconnect (9–10% for Mg thin film$^{39,40}$), as in a previous report$^{37}$, the obtained total stretchability of ~40% corresponded well to the experimental value of Fig. 3e. The strain distribution in the SiO$_2$ layer was similar to that in the Mg layer, although the SiO$_2$ layer would fracture before the Mg layer due to its slightly smaller fracture strain of 8%$^{41}$. In addition, the peak strain of 4.1% in the Si layer was immediately below its fracture strain of 4.25%$^{42}$, which explained the fracture of the Si layer upon further stretching.

**In vitro assessments of dissolvable gas sensors using a semipermeable membrane**

Humidity-free gas meters enable the maintenance of robust performance even in harsh environments in envisioned areas of interest. Figure 4a illustrates the
Fig. 4 (See legend on next page.)
fundamental principle that a PDMS-based semipermeable membrane can effectively deliver desired gas components to active areas of an electronic system and repel water and/or aqueous components. The PDMS membrane provides hydrophobicity and gas permeability since methyl groups (Si–CH₃) on the surface of PDMS reduce the surface energy, and siloxane backbones (Si–O) inside the polymer network form a pathway for NO diffusion. In comparative experiments without (Fig. 4b) or with (Fig. 4c) membranes, absolute values of the responsiveness were slightly reduced, while the electrical characteristics still exhibited a high level of sensitivity compared to those in the tests where there were nearly no responses with even a small amount of moisture in the absence of the membrane. Similar behaviors were found during immersion in aqueous media, in which the membrane enabled stable detection in phosphate-buffered saline (PBS, pH 7.4) solution, whereas no signals were observed without the membrane (Fig. 4d and experimental setup appeared in Fig. S12). Detailed studies of the correlation between membrane thickness and system performance could enhance the sensitivity under optimized conditions (Fig. S13).

In vitro assessment of the NO detection capability indicated the feasibility of using silicon-based bioresorbable devices for applications in temporary implants. Figure 4e briefly describes the overall procedure of NO production that is recognized as a regulator of inflammatory responses in living macrophages. Macrophages undergo M₁ activation in response to pathogen-specific molecules, lipopolysaccharide (LPS) or proinflammatory cytokines such as TNF-α and interferon-γ (INF-γ). As a result of inflammatory activation, macrophages produce inducible nitric oxide synthase (iNOS), one of the key enzymes that acts as a catalyst to produce NO by consuming L-arginine. A set of fluorescence confocal images verified the expression of iNOS, cellular morphological changes (Fig. 4f) and NO production in the presence of L-arginine (Fig. 4g). Quantitative analysis of immunofluorescence intensity is shown in Fig. S14. The produced NO reacted with DAF-FM to increase the fluorescence intensity in Fig. 4h, suggesting that both iNOS expression and the presence of L-arginine were essential factors. Figure 4i shows the measured responses after an injection of L-arginine (150 μM) into control (black) and M₁-activated macrophages (red) in medium (L-arginine-free DMEM (2 ml)). Details of the experimental setup are shown in Fig. S15. The results showed that there was no response in the control group without M₁ activation, while a linear response was clearly seen in the activated group, as expected. Evaluation of the dynamic response over time in Fig. 4j was performed through periodic L-arginine injections and ventilations, verifying the capability of the system for reliable and repeatable measurements of produced NO concentrations.

**Conclusion**

The concepts introduced here provide materials, a manufacturing strategy and device layouts for a soft, transient NO₃ gas detector integrated with temperature and hydration meters. Electrical measurements reveal that the thickness, crystallinity, and operation voltage have a strong influence on the record-high sensitivity and selectivity with rapid response and recovery rates. Upon mechanical deformations by diverse modalities, the performance of large-scale arrays change negligibly, consistent with those determined by theoretical calculations. In vitro studies provide the feasibility of stable operation in harsh conditions, suggesting practical uses in environmental monitoring and biomedical implants.

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