Recent Advances in Cellulose Nanofibers Preparation through Energy-Efficient Approaches: A Review

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Abstract: Cellulose nanofibers (CNFs) and their applications have recently gained significant attention due to the attractive and unique combination of their properties including excellent mechanical properties, surface chemistry, biocompatibility, and most importantly, their abundance from sustainable and renewable resources. Although there are some commercial production plants, mostly in developed countries, the optimum CNF production is still restricted due to the expensive initial investment, high mechanical energy demand, and high relevant production cost. This paper discusses the development of the current trend and most applied methods to introduce energy-efficient approaches for the preparation of CNFs. The production of cost-effective CNFs represents a critical step for introducing bio-based materials to industrial markets and provides a platform for the development of novel high value applications. The key factor remains within the process and feedstock optimization of the production conditions to achieve high yields and quality with consistent production aimed at cost effective CNFs from different feedstock.

Keywords: cellulose nanofibers; energy-efficient; mechanical process; pretreatments

1. Introduction

The use of natural renewable polymers has steadily gained a strong interest over the last three decades and remains as one of the challenges of the 21st century. Cellulose is the most important naturally abundant organic biopolymer in the biosphere and also allocates 1.5 × 1012 tons of the total annual biomass production [1,2]. Cellulose is odorless, has no taste, is hydrophilic, and is insoluble in most solvents including strong alkalis at room temperature. By treating cellulose with concentrated acids known as acid hydrolysis, the amorphous regions can be broken up, thereby producing nanosized cellulose-based crystals called nanocrystalline cellulose (NCC) or cellulose nanocrystals (CNCs) [3]. CNCs are elongated rigid rod-like or whisker shaped particles with a rectangular cross section. CNCs can be prepared from any cellulose source material including wood pulp,
recycled paper, and paperboard, but is not limited to cotton fibers, hemp, flax, bamboo, sugarcane bagasse, and other agro-biomass [4]. Hence, cellulose-based nanomaterials can be extracted from cellulose fibers using physical, enzymatic, and/or chemical treatments [5]. CNFs are materials composed of nanosized cellulose fibrils with a high aspect ratio (length-to-width ratio). Typically, the diameters and lengths are 5–100 nm and several micrometers, respectively [2]. It is a flexible material with crystalline and amorphous regions. CNFs can be produced by a mechanical procedure with or without pre-treatment. The most commonly used approaches to isolate CNF are enzymatic treatment and TEMPO oxidation, followed by mechanical defibrillation. The final product after the delamination process is a gel-like CNF dispersion in water. CNFs have spaghetti-like morphology with typical dimensions of 20–50 nm in width and up to several microns in length (Figure 1).

Figure 1. Top-down approach for preparation of cellulose nanofibers (CNFs) from bagasse.

CNFs are categorized as one of the most promising and emerging bionanomaterials for advanced high-performance bio-nano composite applications [6]. Its discovery has appended exciting innovation into natural plant fiber-reinforced polymer composites [7]. The reasonably low cost of CNF preparation compared to other synthetic and bioinspired nanomaterials raises an excellent opportunity to use them as reinforcing elements in nanocomposites with wide-scale structural applications [8]. In addition, CNFs have a broad range of potential application in different sectors such as bio-based substitution for traditional plastics, reinforcement of composite materials, boosting paper properties, and a barrier material in packaging [5]. The commercialization of CNFs is still suffering from some barriers such as the high cost of raw materials and a high level of energy consumption for the production of CNFs, which needs to be addressed to enable their subsequent applications at the industrial level. Several attempts have recently been reported to reduce the finished-cost of CNFs by using cheaper pristine cellulose fibers and whole lignocellulosic fiber utilization [9–11]. Furthermore, an electrostatically induced swelling by subjecting cellulose fibers to oxidative pre-treatments or grafting procedures, or simply producing pulp with a high number of charged groups have also been proposed as cost-effective
approaches for the preparation of CNFs [12,13]. Along this line, Zimmermann et al. [14] found that wheat straw, recycled paper, and choice of proper pretreatment may reduce the final price of CNF production. The mechanical energy consumption of CNFs with different qualities has been extensively investigated and it was concluded that the selection of optimized pretreatment along with mechanical processing could be a necessity for a cost-effective approach of CNFs compared to petroleum-based plastics in packaging [15]. Extensive chemical pretreatment has also been performed as a route to reduce the energy input required to isolate the nanofibers [16]. Naderi et al. [17,18] used carboxymethyl cellulose as an additive for wood pulp in order to facilitate the microfluidization process and obtain energy-efficient CNFs. Despite the reported attempts, the high energy requirement for CNF production still excludes their entry into the broader commercial market. It is worthy to note that all processes applied for the production of CNFs were not competitive and therefore, the finished cost of CNFs is relatively high. To the best of our knowledge, there is no updated, consolidated, and comprehensive information about the recent development on optimizing the energy consumption of CNF production and introducing energy-efficient processes in the literature. Thus, this review addresses the modifications and innovation in CNF preparation strategies from several standpoints of view such as using cheaper raw materials, applying different effective pretreatments, and efficient-mechanical processes to produce CNFs through energy-efficient approaches.

2. Pristine Fibers

In order to make CNFs commercially viable, their supply chain from fiber resource to extraction processes needs to be optimized. Besides the production cost, considerable attention is required as the morphological properties of CNFs depend on the source of the cellulose, preparation methods, and processing conditions [19]. All lignocellulosic fiber sources vary in chemical constituents as well as the morphology of the plant cell wall (i.e., length, diameter, and cell wall types). In particular, the amount of cellulose, hemicellulose, lignin, and extractives differs not only between the lignocellulosic fibers, but also between various species at different growing geographies. Additionally, cellulose sources differ significantly in their values of crystallinity, fiber length, fiber diameter, and microfibril angle (MFA) [20]. Although woody cellulose fibers are the most popular resource for the production of CNFs, non-woody cellulose fibers are currently being investigated as potential pristine cellulose fibers for the production of varying qualities of CNFs [11]. Facile processing, low energy consumption, vast abundance, low-cost, low lignin content, and renewability are considered as some of the promising features of nonwoody CNFs. Annual agricultural crops such as flax, hemp, kenaf, sisal, cotton, and agricultural residues including sugar cane bagasse, wheat, and rice straw are some non-woody cellulose fibers that have been employed to produce CNFs in the literature [2,21–23]. It is noteworthy that woody cellulose fibrils are mainly found in the secondary cell wall, while cellulose fibrils from agricultural fibers are in the primary cell wall, and thus easier to liberate with less energy demand and less time consumption. In addition, shorter fiber length and thinner cell wall of the non-wood residue compared to wood plant result in processing with no homogenizer clogging, no limitation in inlet consistencies, and low energy demand due to higher feeding consistencies of pristine fibers. Therefore, these low-cost materials are considered as an emerging alternative for the production of CNFs with acceptable properties in an increasingly environmentally friendly circular bio-economy.

2.1. Non-Wood Plant Cellulose Fibers

Zimmermann et al. [14] were the first researchers who used non-wood lignocellulosic fibers such as wheat straw for producing CNFs. They concluded that wheat straw with smaller dimensions (reduced in lengths and diameters) needed a shorter processing time compared to wood-based cellulose fibers through the inline dispersing, and fewer passes through the high-shear homogenizer. In fact, they showed that wheat straw pulp required 30 min for processing via inline dispersing, while beech wood pulp required 60 min for
processing. Later on, Ferrer et al. [24] described the CNF preparation from empty palm fruit bunch fibers (EPFBF) using different sulfur-free pulping systems such as NaOH-AQ, formic and hydrochloric acid (Formosolv), formic acid, and hydroxide peroxide (Milox). The resulting CNFs possessed unique properties favorably comparable with wood CNFs, thus represents a value-added raw material for nanopaper production. In another effort, Hassan et al. [25] investigated the capability of date palm fruit stalks (DPFS) as an underutilized agricultural residue for the production of microfibrillated cellulose (MFC) through enzyme assisted pretreatment. Josset et al. [26] measured the energy consumption of CNF production from wheat straw cellulose fibers. Their results demonstrated that energy consumption increased almost linearly after two grinding cycles. The whole treatment of wheat straw cellulosic fibers consumed 5.75 kWh/kg on a dry-weight base, while bleached softwood-based pulp fibers required 5.25 kWh/kg to liberate. Suopajärvi et al. [27] conducted a competitive study on two different kinds of cellulose fiber sources including wheat straw and birch for the production of CNFs. The produced wheat straw cellulose fibrils were shorter than the birch cellulose, fibrils, and the downsizing of fiber to nanofibrils was easier during the homogenization. In addition to processing costs, the final properties of CNFs strongly depend on the processing techniques, which have been extensively evaluated in the literature. Table 1 summarizes the different mechanical approaches for the production of CNFs from different non-wood plant fibers and their corresponding characteristics.

| Pristine Raw Material | Mechanical Process | Energy Consumption (kWh/kg) | CNFs Diameters (nm) | Reference |
|-----------------------|--------------------|-----------------------------|--------------------|-----------|
| Sugar beet            | Homogenization     | N.D *                       | -                  | [28]      |
| Wheat straw           | Homogenized at pressure ≥300 bar | N.D   | 10–80               | [29]      |
| Soy hull              | Waring blender     | N.D | 20–120              | [30]      |
| Banana rachis         |                    |                             |                    |           |
| Wheat straw           | HPH                | Requires fewer passes through HPH | Homogenous network structures (less than 100) | [14]      |
| Refined fibrous wheat straw | Microfluidization |                       |                    |           |
| Unbleached kenaf bast fiber | HPH     | N.D | 20–40               | [31]      |
| Bleached kenaf bast fiber | HPH     | N.D | 10–30               | [32]      |
| Kenaf core fibers     | HPH                | N.D | 20–25               | [33]      |
| Empty fruit bunches (EFB) of oil palm | HPH | N.D | 5–40               | [34]      |
| Sugarcane bagasse     | Ultafine grinder   | N.D | 5–15                | [34]      |
| Rice straw            | Ultafine grinder   | N.D | 5–15                | [34]      |
| Empty Palm Fruit Bunch Fibers (EPFBF) | Microfluidization | N.D | 4–13                | [24]      |
| Unbleached kenaf pulp | Microfluidization | N.D | 10–30               | [24]      |
| Wheat straw           | Microfluidization | N.D | Less than 100       | [35]      |
| 10 times grinding at 1500 rpm |                   | N.D | 5.75                | [26]      |
| Corn husk             | High intensity ultrasonication | N.D | 50 to 250           | [36]      |
| Triticale crop residue Fiber oxidized with TEMPO | HPH | 40–45 | 20–30               | [37]      |
| Bleached bagasse kraft | Microgrinding      | 33.55 | 23.58 ± 8.63        | [38]      |
| Unbleached soda rice straw | Microgrinding | 11.4 | ≤17                | [22]      |
| Unbleached NSSC rice straw | Microgrinding | 11.2 | ≤17                | [22]      |

* No data available.
More recently, sugarcane bagasse was used as a cheap pristine fiber in CNFs production through micro-grinding with enzymatic pretreatment [38] and no pretreatment [38]. Bleached bagasse fibers consumed a specific total energy of 33.55 kWh/kg for 2 h of grinding to downsize the CNFs, whereas enzymatic pretreatment reduced the specific total energy consumption by 11.83%. In this connection, unbleached bagasse fiber was also used for the production of lignin-rich CNFs through 2 h grinding with an energy consumption of $\simeq 9-10$ kWh/kg. Extending the micro-grinding time to 2 h increased the energy consumption considerably ($\simeq$twofold). In another research work, in general, agricultural by-products are receiving more attention in introducing a commercially competitive route for CNF production since non-wood plant fibers contain less lignin than wood, and their pulping and bleaching processes require lower energy. Other examples of non-wood fibers with potential application in CNFs extraction include those obtained from the cultivation of corn, sorghum, sugarcane, rice straw, barley, pineapple, bananas, coconut crops, and corn husk [3,21].

2.2. Pulp and Paper Mill Sludge

According to the national council for air and stream improvement, the paper industry was the primary producer of water treatment-plant residuals or well-known sludge in the world. It has produced 5.83 million tons of the residuals, in which only 23% was practically employed, while 51% was landfilled or lagoon stored, and another 26% was burned [39–42]. The possible applications of sludge as a cheap raw material in pulp and paper, and the cellulose derivatives industry has been reported. As mentioned earlier, this cheap, accessible, abundant, and sustainable raw material can practically be considered as an emerging raw material for the production of CNFs [33,43,44]. Similarly, Joonobi et al. [33] suggested that sludge of cellulose (dissolving pulp) could be used as an alternative and potential source for scaled-up production of CNFs. They reported that the energy consumption for isolating nanofibers through ultrafine grinding from cellulose and sludge was 1.7 and 1.3 kWh/kg, respectively. Overall, they reported that the grinder with low rotor-speed (1440 rpm) was found to be more energy-efficient compared to the high speed for mechanical fibrillation. In another attempt, Adu et al. [44] conducted comparative research on two different types of pristine fibers including never dried and dried paper mill sludge for CNF production. They found that these starting materials required no specific pretreatment for the production of CNFs. The energy consumption for the fibrillation of cellulose from PSNFD and PSNFD paper mill sludge was found to be 8.5 kWh/kg and 9.3 kWh/kg, respectively. The higher energy consumption in PSNFD was more likely contributed to a well-known phenomenon (i.e., hornification), which usually happens during drying of cellulose and facilitates an irreversible bonding between cellulose fibrils and enhanced degree of crosslinking between cellulose microfibrils due to additionally formed hydrogen bonding [45,46]. It is worth mentioning that pivotal issues such as the variability in sludge chemical composition from different pulp mills should be considered in order to scale-up the production of CNFs [44].

2.3. Secondary Cellulose Fibers

The main (10–39%) ingredients of mixed municipal solid waste are paper and board [47]. Ultimately, the better use of secondary fibers will result in higher resource efficiency, avoiding resource depletion and unnecessary disposal of useful wastes. Nowadays, these undervalued cellulose sources provide a potential of pristine cellulose fibers for the production of CNCs [48] and CNFs [49,50]. Paper produced from repeated recycled fibers is far weaker than a paper made from virgin pulps and can be considered for the production of cellulose nanomaterials. Wang et al. [49] reported that CNFs prepared from an old corrugated container (OCC) via ultrafine grinding followed by ultrasonication had an ultra-long cellulose fibril structure with an average diameter ranging from 30 nm to 100 nm. Josset et al. [26] evaluated the energy consumption and CNF quality of recycled newsprint (RN) and virgin cellulose fibers including wheat straw and bleached wood
cellulose fibers. They deduced that there was a linear relationship between energy consumption and grinding cycles with an energy-consumption rate of ca. 0.7 kWh/kg per grinding cycle (i.e., after 10 cycles of grinding, wood fibers consumed 5.25 kWh/kg, while, wheat straw and recycled newsprint fibers consumed 5.75 kWh/kg and 6.75 kWh/kg, respectively). A different approach was reported by Wang and Zhou [50], who examined mixed office paper (MOP) without deinking for the production of CNFs. They reported that energy consumption for MOP during grinding was 17% lower than fibrillation of bleached kraft eucalyptus cellulose fiber after 3 h. This was more likely attributed to the presence of 18% filler that effectively reduced the number of fibers. This, in turn, reduced the required energy for fiber breakup during the grinding step. Another research group reported the energy consumption of CNF production from different municipal solid waste paper (MSWP) through ultrafine grinding between 6.6 and 13.1 kWh/kg, depending on the purification pretreatment methods applied (i.e., pulping, flotation, and washing) and the number of impurities in the pristine materials [47]. MSWP fiber after pulping and floatation pretreatments represented the lowest grinding time of about 134 min and consumed the minimum grinding energy (6.6 kWh/kg).

2.4. The Effect of Noncellulosic Components on Fibrillation

2.4.1. The Vital Role of Hemicellulose

Generally, the higher the hemicellulose content in the cellulose fibers, the higher the yield of the nanofibrillated material and the lower energy consumption during the mechanical treatment [9,51–54]. Hemicelluloses can surround the cellulose fibrils thanks to their hydrated amorphous character, carboxyl groups, and branched structures. Basically, hemicelluloses behave as a protective colloid, hampering cellulose fibrils and impeding the formation of irreversible hydrogen bonds between cellulose fibrils. The lowest energy consumption (8.2 kWh/kg) was reported for rapeseed stalk cellulose fibers when the mechanical fibrillation process using a waring blender was conducted followed by the D2–O2 (NaClO2 delignification and TEMPO-NaBr-NaClO oxidation) [54]. However, corn stalk and rice straw cellulose fibers consumed 12 and 14.8 kWh/kg, respectively, under the same condition (CNF yield 60%). Furthermore, the application of a high-pressure homogenizer led to an approximately four-fold growth in energy consumption for the same D2–O2 treated cellulose fibers. In particular, rapeseed stalk, corn stalk, and rice straw cellulose fibers consumed 35, 40, and 52 kWh/kg, respectively. The amount of mechanical energy required for the downsizing of hemicellulose-rich pristine cellulose fibers has still not been quantitatively reported.

2.4.2. The Importance of Residual Lignin on the Energy-Efficient Process

The preparation of CNFs from various pristine fibers such as wood, non-wood plant fibers, secondary fibers, pulp, and paper mill sludge depends entirely on fully bleached cellulose fibers. However, unbleached pulps offer a significant improvement in the CNF quality and process effectiveness including total energy-saving and diminishing pollutants loads from bleaching treatments. In this review and hereafter, CNFs produced from unbleached, semi-bleached, and residual lignin-containing cellulose fibers or pulps are referred to as “lignocellulosic nanofibers (LCNFs)”. Heretofore, different types of lignin-containing starting cellulose fibers have been used for the preparation of LCNFs [10,22–24,55–65]. LCNFs are of interest due to their high compatibility with different polymers, huge application in the bio-based packaging sector, and environmentally friendly production, which prevents hazardous substances being released. A focal point of the current section is to explore how the delignification process and residual lignin affect energy consumption during the mechanical fibrillation process. In agreement with a recently published work [66], Ferrer et al. [24] reported that residual lignin content (about ≤1–3%) improved LCNF yield with a smaller pore structure, fibril diameter, higher surface area, and an increased interaction with water due to its charge density, mechano-radical effects (radical scavenging ability), and amorphous structure. The nanopapers prepared
from LCNFs in this study also exhibited better mechanical properties than those with no residual lignin. Jiang et al. [66] reported that the LCNFs containing more residual lignin consumed slightly lower energy during the defibrillation process through the micro-grinding approach. In general, the residual lignin delayed the fibrillation at the initial stage, and if the fibrillation process prolonged further, the cellulose fibers containing more residual lignin displayed the highest fibril yields. Recently, there is also increased attention to applying lignin-containing starting materials for the production of LCNFs [67–70]. The energy consumption for bagasse cellulose fibers with different residual lignin of 14.9%, 2.4%, and 0.1% was measured to be 36 kWh/kg, 37 kWh/kg, and 39 kWh/kg for 7 h of grinding, respectively. The energy consumption of various pristine cellulose fibers after different passes through the grinder was reported as bleached softwood (Radiata Pine) kraft pulp, 13.3 kWh/kg; bleached hardwood kraft pulp (Acacia), 19.34 kWh/kg; hardwood CTMP (poplar), 15.3 kWh/kg; and unbleached softwood (Pinus sylvestris) kraft pulp, 13.3 kWh/kg. The higher energy consumption of bleached hardwood cellulose fibers (19.34 kWh/kg) compared to the softwood (13.3 kWh/kg) was due to the more rigid, complex, and heterogeneous structure of hardwood fibers. He et al. [71] suggested that the low energy consumption of the CTMP cellulose fibers was also due to the highest fines content initially presented, which meant that the exposed cross-sections in fines were more accessible to fibrillation (Figure 2).

**Figure 2.** Field emission electron scanning microscopy (FESEM) images of CNFs, prepared from poplar CTMP (A), unbleached Pine kraft pulp (B), bleached Pine kraft pulp (C), and bleached Acacia kraft pulp (D) [71]. All permissions obtained through Rights Link®.

Overall, three primary mechanisms have been suggested for the importance of the residual lignin on the fibrillation process of LCNFs: (1) the residual lignin could boost the fibril yield via hindering the prepared LCNFs from recombination [66]; (2) the non-etherified phenolic groups of residual lignin (i.e., ortho-methoxy groups, hydroxyl groups, and the double bond between the outermost carbon atoms in the side chain are able of scavenging radicals [10,24,65,67]); and (3) very active radicals of cellulose fibrils due to high density of hydroxyl groups and improved specific surface area could react with lignin nanoparticles and form very stable structures [10,66]. Therefore, the residual lignin in the LCNF structures could remarkably diminish the interfibrillar hydrogen bonding and inhibit the LCNFs from the formation of flocks or bundles [65,66,68,69]. The lignin nanoparticles are clearly distinct in Figure 3A,B (i.e., large amounts of spherical-like lignin nanoparticles (red circles)).
Figure 3. Field emission electron scanning microscopy (FESEM) images of LCNFs (A), prepared from poplar with high lignin content (A, RL = 14.01%), medium lignin content (B, RL = 8.2%), low lignin content (C, RL = 2%), and very low lignin content (D, RL = 0.2%) [71]. Permission obtained through Rights Link® and (B) of LCNFs, prepared from chemi-mechanical (CMP) hardwood pulps (unpublished data). The red arrows and red circles indicate lignin nanoparticles (LNPs).

As can be observed in Table 2, the total energy consumption during the mechanical liberation of lignin-containing starting materials was lower than bleached cellulose fibers. A decisive property of LCNFs compared to CNFs is attributed to a high-water contact angle, which is promising for the potential applications in the sustainable and bio-based packaging industries. Here, it was pointed out that not only is the production of LCNFs from agricultural residues via an energy-efficient process economically possible, but also that its LCNF properties such as tensile strength and modulus of elasticity could surpass those of the wood-based ones.
Table 2. Important characteristics of LCNFs prepared from different pristine cellulose fibers.

| Pristine Fibers                          | Residual Lignin (%) | Type of Mechanical Approach | LCNF Yield (%) | LCNF Diameter (nm) | Energy Consumption (kWh/kg) | Water Contact Angle (°) | Reference |
|------------------------------------------|---------------------|------------------------------|----------------|---------------------|-----------------------------|-------------------------|-----------|
| **Birch kraft pulp:**                   |                     |                              |                |                     |                             |                         |           |
| Unbleached                               | 2.2                 | Grinder                      | -              | -                   | -                           | -                       | [20]      |
| Oxygen delignified                       | 1.5                 |                              |                |                     |                             |                         |           |
| **Birch:**                               |                     |                              |                |                     |                             |                         |           |
| Unbleached                               | 2.7                 | HPF                          | -              | 10–40               | -                           | 60 ± 6                  | [9]       |
| Oxygen delignified                       | 1.9                 |                              | -              | 10–40               | -                           | 54 ± 6                  |           |
| **Softwood kraft pulp:**                |                     |                              |                |                     |                             |                         |           |
| Bleached                                 | 1.0                 |                              | 20             |                     |                             |                         | [21]      |
| Unbleached                               | 2.6                 | HPF                          | 15             |                     |                             |                         |           |
| CTMP                                     | 20.6                | Grinder                      | 9              |                     |                             |                         |           |
| TMP                                      | 25.6                |                              | 2              |                     |                             |                         |           |
| **SEW fiber of spruce:**                |                     |                              |                |                     |                             |                         |           |
| Low lignin                               | 1.7                 | HPF                          | 44 ± 3         | 15.49               |                             | 78                      | [22]      |
| Medium lignin                            | 3.7                 |                              | 20 ± 2         | 15.25               |                             | 78                      |           |
| High lignin                              | 13.5                |                              | 16 ± 2         |                     |                             |                         |           |
| **Wheat straw:**                         |                     |                              |                |                     |                             |                         |           |
| TO-LCNF                                  | 17.7                | -                            | ≥95            | 6.81                |                             |                         | [23]      |
| Mec.LCNF                                 | 17.7                | HPH                          | 55.6           | 14.01               |                             |                         |           |
| Enz. LCNF                                | 17.7                | HPH                          | 37.45          | 14.52               |                             |                         |           |
| **Pine kraft Poplar CTMP:**             |                     |                              |                |                     |                             |                         | [18]      |
|                                          | 6.82                | Grinder                      | -              | -                   | 15.49                       | 15.25                   |           |
|                                          | 18.81               |                              |                |                     |                             |                         |           |
| **Rice straw pulp:**                    |                     |                              |                |                     |                             |                         |           |
| Soda                                     | 16.43               |                              | 13 ± 5         | 11.4                | 61.4                        |                         |           |
| Soda-Enz.                                | 8.06                |                              | 11 ± 5         | 10.5                | 60.3                        |                         |           |
| NSSC                                     | 7.98                | Grinder                      | 17 ± 6         | 11.2                | 69.7                        |                         | [15]      |
| NSSC-Enz                                 | 14.15               |                              | 14 ± 7         | 13.5                | 63.9                        |                         |           |
| **Poplar:**                              |                     |                              |                |                     |                             |                         |           |
| High lignin                              | 22.1                |                              | 15.1 ± 3.5     | 45                  |                             |                         | [24]      |
| Medium lignin                            | 14.1                | Grinder                      | 16.4 ± 3.3     | 41.2                |                             |                         |           |
| Low lignin                               | 2.0                 |                              | 15.4 ± 3.4     | 16.5                |                             |                         |           |
| Very low Lignin                          | 0.2                 |                              | 17.1 ± 3.5     | 9.3                 |                             |                         |           |
| **Organosolv bagasse pulp:**           |                     |                              |                |                     |                             |                         |           |
| High lignin                              | 14.91               |                              | 51             | 9.2 ± 3.3           | 39                          |                         | [19]      |
| Medium lignin                            | 2.41                | Grinder                      | 46             | 22.7 ± 6.7          | 38                          |                         |           |
| Low lignin                               | 0.11                |                              | 39             | 32.9 ± 7.7          | 37                          |                         |           |
| **Kraft bagasse pulp:**                 |                     |                              |                |                     |                             |                         |           |
| Medium lignin                            | 2.7                 | Grinder                      | 46             | 38                  |                             |                         | [14]      |
| Low lignin                               | 0.1                 |                              | 39             | 37                  |                             |                         |           |
| **Hardwood Chemi-mechanical (CMP) pulps:** |                 |                              |                |                     |                             |                         |           |
| High lignin                              | 18.8                | Grinder                      | -              | 19 ± 3.5            | 2.1                         |                         | [25]      |
| Medium lignin                            | 10.9                | Grinder                      | -              | 30 ± 7.3            | 4.5                         |                         |           |
| Low lignin                               | 1.98                | Grinder                      | -              | 40 ± 7.9            | 5.4                         |                         |           |
| **Hardwood neutral sulfate semichemical (NSSC) pulps:** | | | | | | | [25] |
| High lignin                              | 16.8                | Grinder                      | -              | 22 ± 3.8            | 4.2                         |                         |           |
| Low lignin                               | 1.5                 | Grinder                      | -              | 35 ± 6.5            | 5.1                         |                         |           |
3. Mechanical Approaches

Three frequently used mechanical approaches for the production of both CNFs and LCNFs are common refiners, ultrafine supergrinders, or microgrinders, high pressure homogenizers (HPHs), and high pressure microfluidizers (HPMs) (Figure 4). Optimum selection of the aforementioned mechanical techniques relies on decisive factors such as type, morphological, and chemical characteristics of all kinds of the starting materials (Section 1), the fibrillation severity, final product specifications, and the end-use and final applications of the prepared CNFs and LCNFs.

3.1. Microgrinding (MG)

In a grinding process (Figure 4A), the dilute cellulose fiber suspension (usually 1–2.5%) is forced through a gap into the ultrafine grinder where the upper stone is static, and the lower one is rotating. These stones are made of coarse silicon carbide (SiC) and have surfaces fitted with bars and grooves against which cellulose fibers are subjected to repeated cyclic stresses. Immediately after feeding, the rotor speed increases to 1440–1500 rpm to avoid the grinder blocking [21]. The centrifugal forces originating from the repeated cyclic pressure and shearing stress on the fiber suspension result in the fibrillation of the fibers being nanosized in diameter. Jonoobi et al. [33] investigated the efficacy of different rotor speeds (high speed of 3200 rpm and low speed of 1440 rpm) as a critical processing parameter of the grinder on the energy consumption during the production of CNFs from paper sludge. The finding demonstrated a considerable difference in the specific energy consumption between the two rotor speeds where the grinder rotor with the low speed required less energy (1.3 kWh/kg) to downsize the paper sludge compared with that of the high speed (3.1 kWh/kg). This is more likely due to the fact that the cellulose fibers at the lower rotor-speed possess adequate time to act as a lubricating agent between the rotating and static discs, while the higher rotor-speed leads to more contact between the discs, resulting in an increase in temperature and thereby higher energy consumption [33].

3.2. High Pressure Homogenizer (HPH)

Nowadays, the production of CNFs usually consists of a mechanical refining and homogenizing process. For the refining step, both a disk refiner and valley beater are used. The HPH is widely used for the production of CNFs and LCNFs with different qualities at the lab, pilot, and industrial scales (Figure 4B). This technique involves forcing the diluted cellulose fiber suspension (0.5–1%) through a very narrow channel or orifice
using a piston, under a high pressure of 50–2000 bar. The width of the homogenization gap ranging from 5 to 20 \( \mu \)m depends on the viscosity of the suspension and the applied pressure [72]. The created cavitation and shockwaves forces of HPH are powerful enough to downsize the cellulose fibers to CNFs and LCNFs. Nano-sizing the cellulose fibers can be obtained via a sudden pressure drop, high shear force, turbulent flow, and interfiber collisions against each other [73]. The extent of the cellulose liberation relies on the number of passages through the homogenizer and the applied pressure [9,74]. Production of CNFs and LCNFs through HPH now faces various problems including insufficient disintegration of the cellulose fibers, clogging of the homogenizer, high energy consumption, crystalline structure disruption, and the repeated mechanical treatment necessity for achieving high-quality CNFs. To prevail against the aforementioned obstacles, different solutions such as mechanical, enzymatic, and chemical pretreatments have been proposed (refer to Section 3).

### 3.3. High-Pressure Microfluidizer (HPM)

Nowadays, the high-pressure microfluidizer approach is incessantly being applied to various technologies for miniaturization from macrostructure to nano-dimensions [75,76]. HPM is categorized into (a) two-step single-channel microfluidization and (b) one-step dual-channel microfluidization [77]. Among the various mechanical approaches available, HPM has been shown to be one of the most energy-efficient processes for the preparation of CNFs and LCNFs [18,74], all cellulose composites (ACC) [78], and LCNFs [79] with uniform particle size. Generally, during microfluidization, dilute cellulose fiber suspension is forced through a defined fixed geometry with a narrow channel (either Y-type or Z-type) under high pressure (1500–2100 bar) (Figure 4C). Unlike HPH, which operates at a constant processing volume, the HPM operates at constant shear rate, which reduces the likelihood of clogs [75]. Furthermore, the fixed geometry of the interaction chambers (Y and Z type) ensures reproducibility, especially when cellulosic and lignocellulosic fibers are processed multiple times [14,74,80]. The energy consumption of different mechanical approaches including microgrinding, HPH, and HPM were evaluated by Spence et al. [15] for the first time. A significant difference observed for specific energy consumption per pass was reported as 1.1 kWh/kg and 0.17 kWh/kg for HPH and the micro-grinder, respectively. In particular, the HPM energy consumption was measured as 0.056, 0.11, and 0.175 kWh/kg per pass for the applied pressures of 690, 1380, and 2070 bar, respectively. As specified here, it was determined that for HPM, increasing the processing pressure and the number of passes from five to 20 times significantly increased energy consumption. It was concluded that five or eight passes or less were more efficient for HPM and HPH due to the effect of the number of passes on the minimal improvement in MFC film properties. The processing speed rates of cellulose fiber suspension for the production of MFC were measured to be nearly 2 kg/min for micro-grinding, 1 kg/min for microfluidiziation, and 0.2 kg/min for homogenization. Naderi et al. [17] presented an innovative and a process-driven procedure for energy-efficient production of CNFs involving repeated homogenization of the cellulose fiber suspension at low applied homogenization pressures. The energy consumption during the HPM was calculated using the following equation (Equation (1)):

\[
EC_{P,n} \text{ (kWh/kg)} = \frac{(2.2 \times 2 \times p \times n)}{(1600 \times 1.2)}
\]  

where 2.2 kWh/kg is the energy consumed for a 2% (w/w) carboxymethylated CNFs system at one time passing and 1600 bar; \( p \) is the most used homogenization pressure; and \( n \) is the number of passes through the HPM.

The authors employed different pressures and several passes through the HPM. The energy consumption for four passes at 400 bar, four passes at 550 bar, and three passes at 850 bar were recorded and compared with most widely used pressures in the literature (~1700 bar). The results demonstrated a linear relationship between the number of passes through the HPM and the pressure, where enhancing the fraction of nano-sized entities...
increased the storage modulus and the energy consumption. The more passes through the HPM, the more fibrillated the materials, and the higher the energy consumption.

In addition to optimizing the energy consumption, the clogging problem of some mechanical instruments such as HPH and HPM during non-wood fiber processing should be taken into consideration. In order to overcome this drawback, it is necessary to reduce the size of fibers, especially softwood fibers, using several mechanical and chemical pretreatments before passing fibers through the instruments (Table 3). In this regard, bio-residue fibers such as agricultural residues (i.e., sugarcane bagasse [5,9] and wheat straw [81]) can be considered as efficient pristine materials due to a shorter fiber length, finer fiber diameter, and no need for a supplementary pretreatment.

Table 3. The processing conditions of the mechanical approaches for the preparation of CNFs.

| Type of Mechanical Process | Initial Consistency of Starting Material (%) | Fibrillation Systems | Necessity of Pre-Treatments | Output (kg/min) |
|---------------------------|--------------------------------------------|---------------------|-----------------------------|-----------------|
| Ultrafine grinding        | 1–2.5                                      | Disk grinding       | No                          | 2               |
| HPH          | 0.5–1                                      | Homogenization      | Compulsory; one of the mechanical, enzymatic, and chemical pre-treatments | 1               |
| HPF          | 1                                          | Fluidization        | Compulsory; one of the mechanical, enzymatic, and chemical pre-treatments | 0.2             |

3.4. Twin Screw Extruder (TSE)

Today, there has been an increasing interest toward energy-efficient approaches for the preparation of CNFs. For the first time, a twin-screw extruder (TSE) was used as an emerging fibrillation method for the preparation of CNFs by Ho and co-workers [82]. The bleached kraft pulp was extruded at the concentration of 28 wt%, and it was reported that this kind of pulp could successfully be disintegrated into high-quality fibrillated cellulose fibers (ca. 33–45 wt% solid content). TSE, as a top-down approach in producing CNFs, showed better quality, higher inlet consistency, and lower energy consumption compared to HPH, HPF, and micro-grinding [82–86]. Baati et al. [83] reported that energy consumption during CNF fibrillation with TSE was estimated to be about 4.1 kWh/kg, which a lower value than that required for HPH (between 30 and 50 kWh/kg) or even for micro-grinding (between 5 and 30 kWh/kg). The energy requirement for the production of CNFs in TSE was 63% lower than for producing CNFs from same starting material via the micro-grinding process [86]. Similar findings were reported for TSE and HPH, where the combination of one pass through TSE and two passes through HPH decreased the energy consumption by 45% [87]. The produced CNFs had a similar quality to CNFs produced after five passes through the homogenizer alone. Interestingly, passing the starting material once through the TSE increased the HPH feeding consistency from 2 to 4 wt%. More recently, CNFs were successfully prepared from enzymatic and TEMPO-oxidized cellulose fibers using a twin-screw extruder equipped with kneading disks and fully flighted conveying screws (Figure 5) [84]. The produced CNFs were of high inlet consistency (20–25 wt%), and the method consumed 60% less energy compared to conventional processes.
Figure 5. Different sections of a twin-screw profile for the preparation of CNFs [87].

Overall, TSE, as an emerging mechanical approach, is easy to implement without clogging during the disintegration process. It can be scaled up for high capacity and cost-effective production of CNFs at high consistency. However, optimization of the screw profile and other required TSE processing conditions to achieve the sufficient shear forces necessary for fibrillation of the fibers along with preventing the cellulose degradation remain challenges to overcome.

4. Pretreatments

4.1. Enzymatic Pretreatment

Many researchers have stated that well-documented enzymatic pretreatment decreased the energy consumption during the fibril liberation through HPH, HPM, [19,69,88–98], and micro-grinding [38,90]. Table 4 addresses different enzymes such as mono-component endoglucanase, and xylanase which have been employed to reduce the size and pre-defibrillate the fibers, thus reducing the frequency of equipment clogging. The use of enzymatic pretreatment not only reduces the processing cost by reducing the number of passes through the homogenizers, but also favors environmental efficiency in comparison with chemical methods. Pääkkö et al. [69] studied the application of mechanical shearing without enzymatic hydrolysis. Their results confirmed the preparation of non-homogeneous materials due to severe blocking during the homogenization step. Delgado-Aguilar demonstrated that enzymatic pretreatment prepared higher amounts of fibrils (69%) compared to sole mechanical treatment (21%). A set of enzymes used for the pretreatment of cellulose fibers were well-categorized by Henriksson et al. [88]:

A. Cellobiohydrolases: A and B type cellulases that attack the crystalline structure.
B. Endoglucanases: C and D type cellulases for breaking the amorphous region.
C. Lytic polysaccharide monooxygenase (LPMO).

Table 4. Different enzymes employed to pretreat the cellulose fibers on various raw materials before mechanical fibrillation for the production of CNFs.

| Starting Material       | Used Enzyme           | Process                    | Results                                                                 | Reference |
|-------------------------|-----------------------|----------------------------|-------------------------------------------------------------------------|-----------|
| Bleached softwood sulfite | Endoglucanase       | HPH                        | High aspect ratio, the stronger gel network                             | [28]      |
| Bleached softwood sulfite | Endoglucanase       | HPH                        | High aspect ratio, lower DP, Homogenous distribution of nanofiber       | [27]      |
| Bleached kraft bagasse  | Cellulase, Xylanase  | Grinding and homogenization | Smaller diameter microfibrils bands, reducing energy consumption        | [8]       |
| Bleached softwood pulp  | Cellulase            | Grinding                   | Improving the efficiency of fibril production, reduction in operating cost | [29]      |
| Bleached kraft eucalyptus pulp | Cellulase          | Microfluidizer             | Facilitating mechanical homogenization, reducing the DP                 | [30]      |
| Bleached soda date palm fruit stalks | Xylanase          | High-shear ultrafine friction grinder | Facilitating mechanical homogenization, reducing the DP                 | [31]      |
| Bleached soda bagasse   | Endoglucanase       | HPH                        | Better fibrillation                                                     | [32]      |
Table 4. Cont.

| Starting Material      | Used Enzyme                                      | Process                  | Results                                           | Reference |
|------------------------|--------------------------------------------------|--------------------------|---------------------------------------------------|-----------|
| Banana peel            | Xylanase                                         |                          | Higher aspect ratio, higher crystallinity         | [33]      |
| Bleached kraft hardwood| Novozym® 476                                     | HPH                      | Higher fibrillation                               | [34]      |
| Soybean straw          | Enzymatic cocktail (xylanase, endoglucanase)      | Turrax and sonication    | The smaller diameter of nanofiber                 | [35]      |
| Bleached kraft eucalyptus pulp | Purified and commercial Endoglucanase (Ph-GH5) | Microfluidization        | Nanosize uniformity of CNFs, DP reducing         | [16]      |
| Bleached softwood pulp | Mono-component endoglucanase                     | Microgrinding            | Mechanical energy-saving, uniformity of cellulose fibrils | [36]      |

Endo-cellulase randomly cleave internal hydrogen bonds at amorphous sites that create new chain ends. Additionally, Jeoh et al. [91] concluded that endo-cellulases hydrolyzed hydrogen bonds randomly at amorphous regions within a chain, which yielded shortened average chain lengths and broadened chain length dispersity at the surface with limited impact on total chain populations. Exo-cellulases cleave two to four units from the ends of the exposed chains produced by endo-cellulases, resulting in groups such as cellobiose [92] (Figure 6).

![Figure 6. Proposed reactions of different cellulase enzymes on cellulose fibers [93].](image)

Using a purified and well-characterized endoglucanase can eliminate unknown constituents in commercial endoglucanase, which may affect fibril length shortening. By using a selective enzyme (no reaction to xylan and xyloglucan, and low activity with soluble cello-oligosaccharides), it was shown that the purified enzyme pretreatment could reduce mechanical energy input. The energy consumption decreased at least 30% in comparison with prepared fibers with no enzymatic pretreatment. A recent study showed that the commercial endoglucanase enzyme was more aggressive in reducing DP and released fewer fine fibers compared to purified ones [99,100]. The CNFs through ultrafine grinding
appeared to be less homogeneous in terms of fibril diameters, while in CNFs produced by high-pressure homogenization or microfluidization, large bundles of fibrils were observed [90]. On the effectiveness of post-treatment of the fibers on the production of CNFs, Wang et al. [90] reported that endoglucanase post-treatment efficiently improved the nanosize uniformity of CNFs from ultrafine grinding. More than 80% of the resultant CNFs had a diameter distribution between 5 and 9 nm. The aforementioned treatment could be considered as an effective way for the production of nanosized uniform CNFs. Interestingly, commercial endoglucanase pretreatment at 0.01 mg/g cellulose fiber was much more effective than post-treatment under the same dosage for the production of CNFs with uniform fibril diameters. Commercial endoglucanase contains exoglucanase that may aggressively attack crystalline cellulose and lead to hydrolysis of disordered cellulose regions. However, purified ones may primarily work on non-crystalline regions of cellulose. In addition, xylanase enzyme generally attacks the amorphous regions, helps to cleave the covalent bond such as β-1-4 glycosidic linkage of xylan, and hydrolyzing part of the hemicellulose [90]. On the other hand, cellulases can access the amorphous region of cellulose by endoglucanase function or the extremity of the chain by the action of exoglucanases. Worth mentioning is the fact that no enzyme pre-treatment approach has led to extensive encumbrance and high energy consumption during HPH, illustrating the necessity of enzymatic pretreatment through cellulose fiber homogenization. Enzyme dosage is an important aspect of pretreatment. Interestingly, CNFs produced from a combination of enzymatic pretreatment, and high shear homogenization exhibited long fibrils that led to the formation of an entangled network structure [5]. A 3 wt% enzymatic CNFs translucent gel with shear thinning behavior exhibited a storage modulus (G') of 104 Pa [101]. The high viscosity of CNFs can be an interesting characteristic for some applications such as a rheology modifier in food or cosmetics [69]. It should be mentioned that the CNFs viscosity recovers over time when the shear forces are removed. Recently, the Technical Research Center (VTT) in Finland has developed HafCel® technology known as a low-cost method (energy cost ~1 €/ton) for producing high consistency CNFs with low energy industrial mixers. This technology uses the commercially available cellulase enzyme, while the fibrillation degree can technically be adjusted by treatment time and applied enzyme concentration. The process yields very high consistency CNFs (tenfold increase in the solids content, i.e., 20–40%), whereas conventional CNF production methods usually yield 1–3% consistency.

4.2. Chemical Pretreatments

4.2.1. TEMPO-Mediated Oxidation

In recent years, to decrease the mechanical energy required to downsize the cellulose structure through HPH and HPM, chemical pretreatment of cellulose fibers has become popular. This route converts the hydroxyl groups (-OH) of cellulose fibers into carboxylate groups (COO−), thus creating anionic charges that subsequently turn into repulsive forces within the internal structure of the fibers (Figure 7). This also weakens the cellulose structure, which then facilitates the subsequent mechanical disintegration of the fibers to fibrils. In this direction, one of the preferred and well-documented pathways is the 2,2,6,6-tetramethylpiperidine-1-oxyl (TEMPO) mediated oxidation with sodium hypochlorite and bromide and chlorite salts (TEMPO/NaClO/NaBr or TEMPO/NaClO₂/NaClO) as the most common oxidizing agents [2,12,16,52,102–110]. The hydroxyl groups on the C6 of glucose units are converted to a carboxylic group. The nitroxy radicals and nitrosonium salts as an oxidative route transform hydroxyl function into carboxyl and/or aldehyde groups [16,111–113].
Figure 7. The most-applied approach of TEMPO-mediated oxidation of cellulose primary hydroxyl group to carboxyl groups.

Chemical modifications reduce the liberation energy of cellulose fibers from the order of 100 kWh/kg for unmodified cellulose preparations to as low as 1–2 kWh/kg [2,16,114,115], depending on the extent of the treatment [16,115,116]. The discovery of using TEMPO-mediated oxidation as an energy-efficient pretreatment was introduced by Professor Akira Isogai, Associate Professor Tsuguyuki Saito, and Dr. Yoshiharu Nishiyama and they have honorably received the Wallenberg prize for this. This research group motivated more intensive research on CNFs and their applications in different fields. Through this method, the energy need reduced from 30,000 kWh/ton to 100–500 kWh/ton, and the CNFs produced were more homogeneous than that of the enzymatic and no pretreated cellulose fibers. TEMPO-mediated oxidation was also applied to various kinds of native celluloses both wood [96,102,115,117,118] and non-wood cellulose fibers [116,119]. Okita et al. [12] reported that TEMPO-mediated oxidation (TEMPO/NaBr/NaClO system) procured more than 90% of the oxidized groups as a sodium carboxylate with zeta potentials of approximately −75 mV for different native cellulosics. TEMPO-mediated oxidation converted almost all C6 primary hydroxyls on the surface of cellulose microfibrils to sodium carboxylate groups (approximately 1.7 groups/nm²). Delgado-Aguilar et al. [96] evaluated the energy consumption of bleached kraft hardwood pulp (BKHW) through TEMPO-mediated oxidation and reported that TEMPO-oxidized fibers required fewer passes through HPH than other types of pretreatments. In this study, the mechanical energy required was approximately 4–7 kWh/kg [96]. Regarding energy consumption, Isogai et al. [115] reported that TOCN (TEMPO-Oxidized Cellulose Nanofiber) with 90% yield required <1.94 kWh/kg for mild disintegration in an aqueous medium. The produced CNFs with a uniform width of 3–4 nm, and length <2–3 µm proved to be promising materials for potential applications in packaging and display (e.g., high gas-barrier films), fine separation filters, health care materials, and nanofibers for composites. In general, in TEMPO treatment, the addition of more NaClO resulted in an increase in the density of carboxylate groups on the surface of the CNFs and the oxidation time [103]. With an increase in the amount of NaClO from 3.8 to 5.0 mmol/g, the carboxylate content increased from only 0.2 to 0.3 mmol/g, and the oxidation time increased from 40–45 min to 115–130 min. Since, the most commonly used and ongoing chemical pretreatment is TEMPO-mediated oxidation, Table 5 presents the well-documented strategies that have been published thus far.

Table 5. Most-cited TEMPO-mediated oxidation pre-treatments addressed in the literature.

| Starting Materials               | TEMPO System Used         | Fibril Diameter (nm) | Carboxylate Group Content (mmol/g) | Reference |
|---------------------------------|---------------------------|----------------------|-----------------------------------|-----------|
| TEMPO-oxidized spruce holocellulose | TEMPO, NaBr, NaClO       | Crystal size (nm)3.2 | 1.16                               | [37]      |
| TEMPO-oxidized cotton linters   | TEMPO, NaBr, NaClO       | 6.2                  | 0.67                               |           |
| TEMPO-oxidized ramie            | TEMPO, NaBr, NaClO       | 5.8                  | 0.94                               |           |
| Never-dried hardwood            | TEMPO, NaBr, NaClO       | 3–4                  | 1.5                                | [38]      |
| Once-dried hardwood             | TEMPO, NaBr, NaClO       | 3–4                  | 1.5                                |           |
| Softwood kraft pulp             | TEMPO, NaBr, NaClO       | 3–4                  | 1.5                                | [39]      |
| Hardwood kraft pulp             | TEMPO, NaBr, NaClO       | 3–4                  | 1.5                                |           |
Table 5. Cont.

| Starting Materials       | TEMPO System Used                  | Fibril Diameter (nm) | Carboxylate Group Content (mmol/g) | Reference |
|-------------------------|------------------------------------|----------------------|-----------------------------------|-----------|
| Hardwood kraft pulp     | TEMPO, NaClO$_2$, NaClO            | 5                    | 0.8                               | [40]      |

4.2.2. Carboxymethylation

Carboxymethylation pretreatment is one of the most prevalent pretreatment routes for the production of CNFs, where anionic carboxylate groups are introduced onto the cellulose fibers [2,13,80,120–126]. The attachment of negative charges provided by monochloroacetic acid through the controlled reaction condition (Figure 8A) creates an electrostatic repulsion that individualizes the cellulose fibers into cellulose fibrils in the following mechanical treatment process.

Figure 8. Carboxymethylation pretreatment of cellulose with monochloroacetic acid (A). Periodate-chlorite sequential oxidation pretreatment of cellulose fibers (B).

Ankerfors et al. [101] estimated an electricity consumption of 2.33 kWh/kg for treatment through microfluidization after carboxymethylation pretreatment. Naderi et al. [18] introduced a novel protocol using different passing of cellulose fiber suspension at varying applied pressures of 400, 550, 850, and 1700 bar through the homogenizer. They concluded that this route led to more homogeneous and impressible fibrillation of cellulose fibers compared to one-time homogenization through HPM, which is currently accessible. In this regard, the energy consumption of cellulose fiber increased from 0.925 to 3.7 kWh/kg during one to four passes through the HPM at 400 bars. However, three passes of homogenization of cellulose fiber through HPH at 550 and 850 bar, and one step homogenization at 1700 bar required 3.8, 5.9, and 3.9 kWh/kg, respectively [17]. Energy consumption of different cellulose nanofiber pretreatment systems such as enzyme and carboxymethyl pretreatments through microfluidizer (M-110EH, Microfluidics Corp., Westwood, MA, USA) was comprehensively studied by Naderi et al. [17] The energy consumption of enzymatically pretreated CNFs produced from sulfite dissolving pulp was shown to be 2.5 kWh/kg in comparison with carboxymethylated CNFs, which presented 2.4 kWh/kg at a similar passing (one passing) and the same consistency (2%). It was postulated that enzymatically pretreated CNFs showed a lower degree of fibrillation and better barrier properties compared to carboxymethylated CNFs, while carboxymethylated CNFs displayed superior tensile strength, lower residual fiber fragment, and higher shear viscosity than enzymatically pretreated CNFs [17]. Conversely, according to a life cycle study by Arvidsson et al. [125], chemical pretreatment does not necessarily decrease the required energy consumption during the CNF preparation. The current study concluded that carboxymethylation pretreatment, due to the high input of chemicals such as ethanol,
isopropanol, and methanol, addresses a higher cumulative energy demand compared to the homogenization approach only [125]. In this case, the environmental impacts of carboxymethylation pretreatment would be higher, albeit less mechanical energy usage is needed.

4.2.3. Carboxymethyl Cellulose (CMC)-Modified Cellulose Fibers

Irreversible attachment of CMC on cellulose fiber has initially been derived from its applications in paper [127–130]. Ankerfors et al. [101] was the first group to propose this pretreatment, and they examined three types of CMCs including one amphoteric and two anionic. The amphoteric CMC enabled high attachment level of up to 113 mg/g with charge densities of up to 231 µeq/g. However, by using a low charged anionic CMC, an attachment level of up to 62 mg/g was achieved. The mechanism behind the effect of CMC attachment as a pretreatment for cellulose nanofiber production was theoretically explained as follows. The CMC introduces charged groups that swell the fibers, however, it also serves as a lubricant for the fibers due to the electrostatic repulsion. The amphoteric CMC was more effective compared to anionic CMC in terms of lower CMC attached. It was concluded that the higher consistency of CMC-modified pulp resulted in lower energy consumption: CMC-modified pulp with 0.13% consistency consumed 36.3 kWh/kg for one passing and 181.5 kWh/kg for five passings through the HPH. However, increasing the initial consistency to 1.73% required 2.7 and 13.5 kWh/kg energy for one and five passing, respectively [18]. Naderi et al. [131] showed that negative charges produced on the fiber surface through carboxymethylation pretreatment resulted in an efficacious swelling of the fiber, and the fibrillation process was more effective than the enclosing of CMC on the cellulose fiber surfaces. Considering that the carboxymethylation pretreatment is required to be performed in a non-aqueous medium, (e.g., ethanol or isopropanol) to be efficient, the process is relatively complicated as several solvents must be applied and the solvent recovery is relatively pricy, and efficient delamination of fibers requires more energy. Therefore CMC-modified CNFs through an aqueous method, is potentially interesting for applications that do not require highly fibrillated CNFs such as a reinforcing agent in paper and cardboard materials, a rheology modifier in food products, and in the production of bio-based barrier materials [130].

4.2.4. Periodate–Chlorite Oxidation

Periodate–chlorite as a sequential regio-selective oxidation pretreatment has recently been performed for the production of CNFs. This pretreatment reduces the hydrogen bond by increasing the anionic charge density of cellulose fiber surfaces, thereby decreasing the required mechanical energy for the liberation of cellulosic fibers [61,85,132–136]. Briefly, cellulose fiber was mixed with NaIO₄ at 55 °C, filtered and reacted with NaClO₂ for two days at room temperature in 1 M acetic acid. After adjusting the pH to 7.5 with sodium hydroxide, the cellulose suspension at 0.5 wt% was homogenized via one-to-four passes through a double chamber of HPF. As shown in Figure 8B, sodium meta-periodate oxidizes the hydroxyl groups of cellulose at positions 2 (C2) and 3 (C3) to produce aldehyde groups and subsequently oxidize the aldehyde groups to anionic carboxylate groups using sodium chlorite. Periodate–chlorite oxidation pretreatment converts aldehydes partially to anionic carboxylate groups. This would afterward promote the fibrillation of fibers and break the cellulose chains to some extent, despite its selectivity. Broken cellulose chains lead to a decrease in DP [134,136] and crystallinity of the cellulose fibers [61,137], which is desirable for reducing the energy consumption during the homogenization. Along this line, the DP of Pinus radiata market cellulose fiber pretreated with periodate oxidation pretreatment was reported to be <70. It was also shown that the highly oxidized cellulose fibers (carboxyl contents of 1.20 and 1.75 mmol/g) after periodate oxidation presented a transparent appearance, whereas the samples with lower carboxyl contents maintained their fibrous structures. Previously, this type of pretreatment is followed by other reactions such as sulfonation [138,139], reduction with sodium tetrahydridoborate [140,141], and metal salts
(i.e., LiCl and CaCl$_2$) as activators with milling treatment [132] for better fibrillation of different cellulose fibers. This type of pretreatment is very complex to implement in the industry despite all of its advantages and its applicability in different reactions.

4.2.5. Emerging Pretreatments for Production of CNFs

Phosphorylation

It is well known that introducing charged groups into cellulosic fibers reduces the strong cohesive forces among the fibrils, disrupts the cooperative hydrogen bonding [142–144], facilitates the fibrillation of pristine cellulose fibers, and subsequently decreases the energy consumption of the mechanical step and produces stable CNFs with excellent functional properties. Phosphorylation is currently categorized as one of the novel pretreatments, capable of providing negative charges on the cellulose fibers [131,145,146] and subsequently establishing better fibrillation. Briefly, the cellulose fibers were impregnated with (NH$_4$)$_2$HPO$_4$ and urea at 1 wt%, then filtered into 10 wt%, dried at 70 $^\circ$C, and then cured at 150 $^\circ$C for different durations. Interestingly, the phosphorylated CNFs prepared from passing 1–3 times through HPH at 1650 and 1700 bars were easily individualized to a cellulose fibril with 3 nm and 10 nm, respectively. In addition, Noguchi et al. [146] prepared cellulose fibrils with 3–4 nm at 250 MPa through Star Burst Labo. The corresponding prepared nanopapers demonstrated outstanding mechanical properties comparable to those of nanopapers prepared by Henriksson et al. [19]. The phosphate groups attached to the cellulose fibril surfaces provided a self-extinguishing behavior and improved thermal stability compared to those of the unmodified cellulose fibers. The prepared phosphorylated CNFs exhibited higher barrier performance and lowered oxygen permeability value (at RH 50%) than the fibrillated CNFs (i.e., TEMPO-mediated oxidation-based CNFs and carboxymethylated CNFs). It is noteworthy that the pretreatment, unlike other pretreatments, did not significantly change the intrinsic properties of CNFs such as viscosity-average DP and crystallinity index [146]. The nanopapers prepared from the phosphorylated CNFs showed extremely high transparency and represented a natural cellulose-based flame-retardant material. These unique characteristics of the phosphorylated cellulose nanopapers are promising for potential high value-added applications. Hence, this pretreatment proves that phosphorylated CNFs are able to establish a unique application of cellulose nanofiber, which is most interesting from an industrial point of view. Along this line, Oji Holding Corporation and Betulium applied this pretreatment as an attractive alternative to TEMPO-mediated oxidation due to its easy implementation and non-toxic approach. It should be mentioned that there is no information regarding the mechanical energy consumption of this pretreatment in the literature.

Sulfoethylation

Sulfoethylation can be specified as an emerging pretreatment for the production of CNFs with more attractive characteristics compared to conventional ones. Naderi et al. [142] proposed a pretreatment process called the Michael reaction where never-dried cellulose fiber was dispersed in water at 10,000 revolutions using a laboratory pulper. The cellulose fiber (16 g dry mass) was soaked three times in 2-propanol (300 mL) using a kitchen blender and filtered off. The wet cellulose fiber was transferred into a flask, and then the solvent was exchanged with 2-propanol (300 mL). A solution of sodium hydroxide in half of the required amount of aqueous sodium vinyl sulfonate solution (VS, 25% (w/w) was added dropwise at room temperature. The temperature increased to 80 $^\circ$C, while stirring. After one hour, the remaining amount of VS was added to the mixture and stirring was continued for an additional three hours. The mixture was afterward poured into methanol, neutralized with acetic acid, washed with methanol/water (70/30, v/v), and filtered off. In the mentioned work, there was no need for any pretreatment such as periodate oxidation, which is often undesirable for CNF preparation, yet complicated in industrial implementation [87]. In another study, sulfoethylated CNFs were prepared by mechanical treatment via a microfluidizer (one pass at 1700 bar through two Z-shape
200 µm and 100 µm chambers). It was pointed out that this emerging pretreatment did not reduce DP, but crystallinity decreased from 61% to 41% [142]. Of interest to the conclusions is that this emerging pretreatment, due to the characteristics of its attached functional groups (−CH$_2$CH$_2$SO$_3$−Na$^+$), exhibited superior properties such as better fibrillation, good-quality CNF lengths (≤1 µm) and widths (≈5 nm), good dispersibility, higher stability at different pH, and better water solubility compared to carboxymethylated CNFs. Additionally, the prepared nanopapers from sulfoethylated CNFs showed excellent barrier properties rather than the carboxymethylated CNFs.

Deep Eutectic Solvent (DES)

Deep eutectic solvents (DESs) have been proposed as an emerging pretreatment for the preparation of CNFs. Although DESs can be prepared through non-ionic species, they are often classified as ionic liquids (ILs). Generally, DESs are ambient temperature ionic liquids composed of two or three ionic compounds comprising of a hydrogen bond transferor (HBD) such as urea, glycerol, and ethylene glycol along with a hydrogen bond catcher (HBR) capable of combinations to form a eutectic mixture [147,148]. A lower melting point of DESs than each individual component is a peculiar property as the strong hydrogen bonding prevents crystallization of parent components [149]. DESs possess similar physico-chemical properties in comparison to ILs, however, they are considered as an inexpensive, eco-friendly, biodegradable, and green solvent with relatively low toxicity [150–152]. Recently, various urea-based DESs including choline chloride (ChCl) [152–154], ammonium thiocyanate, or guanidine hydrochloride [149] have been successfully employed for the pretreatment of cellulose for the production of CNFs. For example, Suopajärvi et al. [151] pretreated 25 g of different secondary cellulose fibers at a consistency of ∼30% with the DESs solution (choline chloride and urea). They mixed the components for two hours at 100 °C, washed the ensuing suspension, and used a Masuko super masscolloider MKCA6-2 (Kawaguchi, Japan) grinder (consistency of 1.5%) or a Microfluidics M-110EH-30 (Westwood, MA, USA) microfluidizer to obtain CNFs (consistency of 0.4%) through various chambers, passes, and pressures of high-pressure homogenization. The homogenized pulp had a gel-like appearance with shear thinning behavior and a web-like nanofibrous structure. Although it was concluded that the mechanism behind the DESs reaction with cellulose was unknown, Sirviö et al. [153] theorized that the DESs solvent could penetrate the cellulose fibers and loosen their structure, allowing for more efficient fibrillation. They also proposed a possible assumption that a small number of hydroxyl groups are converted to carbamates, which led to an extent, to a similar distortion of the hydrogen bonding network such as TEMPO-mediated oxidation or carboxymethylation [153]. More recently, Tenhunen et al. [152] suggested that cationic choline ions interacted with the anionic groups of cellulose fibers via electrostatic interactions. In this context, Table 6 summarizes different DESs applied for pretreatment of cellulose and corresponding CNF characteristics.

Cellulose fibers were observed to disintegrate into cellulose nanofibrils with widths of 2–21 nm during the DESs pretreatment. In this context, the fiber morphology had not been altered, and negligible changes were noted in the carbohydrate composition [152,153]. These findings suggest that the used DES systems were capable of swelling and loosening the fiber macroscopic structural integrity without notable degradation of the molecular cellulose chain structure. One limiting factor in the DES system, in light of all the concluded advantages, is the prolonged reaction time for choline chloride and urea (16 h at 100 °C [152], 15 h at 90 °C with sodium hydroxide [150], and 2 h at 100 °C [151]). Overall, from Table 6, it can be observed that this novel pretreatment does not significantly decrease the fibers’ DP and leads to CNFs with acceptable transmittance. DESs are classified as a non-hydrolytic pretreatment that can enhance the fibrillation of cellulose fibers and are also categorized as a green, easy-to-use, inexpensive, and recyclable pretreatment; however, their use still requires further research. There is no comprehensive information regarding the advantage
of this pretreatment on energy consumption during the mechanical treatment for the production of CNFs in the relevant literature.

Table 6. Different DESs used for the pretreatment of cellulose and corresponding CNF characteristics.

| Pristine Fibers                        | DESs System Used                  | Applied Mechanical Treatment | CNF Properties | References |
|----------------------------------------|-----------------------------------|------------------------------|----------------|------------|
| Bleached birch pulp (Betula pendula)   | Microfluidizer                    |                              |                | [45]       |
| Bleached birch pulp (Betula pendula)   | Microfluidizer                    |                              |                | [48]       |
| Commercial softwood dissolving pulp    | Microfluidizer                    |                              |                | [49]       |
| Waste board (WB) and milk container    | Masuku grinder                    |                              |                | [46]       |

4.2.6. Other Pretreatments

Ozone

Some unconventional pretreatments can also be applied to cellulosic fibers for the production of CNFs. Ozone gas, along with syringic acid, was used to enhance the oxidation reactions of enzymatic pretreated cellulose fibers [155]. Briefly, syringic acid powder (0.25 g) was mixed with 150 mL of 1% w/v enzyme-treated pulp, and the pH of the resultant solution was increased to 11 using 1 M NaOH to enhance the reactivity of the cellulose hydroxyl groups. Ozone-containing gas was subsequently exerted at a flow rate of 0.25 l/min with a concentration of 18 mg/l (1.26 wt.%). The pulp suspension was persistently stirred on a stir plate, and in a fume hood at room temperature (20 °C). The efficiency of the current pretreatment was compared to the conventional TEMPO-mediated oxidation process. It was found that ozone oxidation accompanied by a lignin-derived phenolic (syringic acid) could be nominated as an alternative to TEMPO-mediated oxidation of the cellulose surface to produce CNFs [155]. The ozone gas alone was not able to oxidize the cellulose fiber surface through reaction with free hydroxyl groups, but the hydroxyl radicals can also form during the reaction of ozone along with syringic acid with cellulose fibers. Similar to TEMPO-mediated oxidation, these radicals relatively oxidized the primary alcohol located on the C6 of the glucose ring to stable carboxylate groups. Interestingly, in contrast to TEMPO-mediated oxidation, syringic acid can be described as a natural oxidant degraded during the oxidation reaction. The CNFs morphology prepared from this extraordinary pretreatment was explored to be 3–10 nm widths and lengths >100 nm after ultra-sonication of the oxidized product in an aqueous suspension. The prepared cellulose nanofibers through this pretreatment had a carboxylate content such as conventional carboxylated cellulose prepared by TEMPO-mediated oxidation.

Fenton Pretreatment

Fenton pretreatment is defined as an acidic hydrogen peroxide pretreatment in the presence of ferrous ions (Fenton’s reagent). Fenton oxidation is a reaction of ferrous ion (Fe²⁺), catalyzing hydrogen peroxide (H₂O₂) to produce hydroxyl radicals (HO·). Hydroxyl radicals, as a strong oxidant, can oxidize and degrade organic polymers such as cellulose. It results in an increase in total charge and number of carbonyl groups, lowering intrinsic viscosity [23]. Accordingly, the Fenton pretreated cellulosic fibers are easier to homogenize
compared to enzymatic pretreatment (i.e., lower viscosity (190–280 dm$^3$/kg) and DP (450–660), higher surface area charge, and higher carbonyl content (79–105 µmol/g)) at a constant homogenization time. It was shown that the Fenton pretreatment yielded a higher degree of fibrillation compared to enzymatic pretreatment for the preparation of MFC after similar passes through the pilot plant HPH. In addition, Fenton pretreatment eventuated in slightly lower energy after 2–5 passes through the HPH. In fact, the energy required for five passes through HPH was reported to be 10 kWh/kg for Fenton pretreatment and 12 kWh/kg for endoglucanase enzymatic pretreatment [156]. Therefore, this eco-friendly pretreatment is considered as a promising alternative to enzymatic pretreatment for the preparation of MFC. It can also provide an opportunity to introduce an energy-efficient pretreatment to achieve a better quality of MFC in terms of the number of fibril elements.

**Oxone® Pretreatment**

The Oxone® pretreatment was recently accredited as a novel, green, and one-pot oxidant pretreatment for cellulose pulp fibers [157,158]. In general, Oxone® is the trade name of the stable triple salts with the composition of 2KHSO$_5$·KHSO$_4$·K$_2$SO$_4$. The active component potassium monopersulfate (KHSO$_5$, potassium peroxomonosulfate) is salt from Caro’s acid H$_2$SO$_5$. The use of Oxone® has rapidly increased, thanks to its stability, simple handling, non-toxic nature, the versatility of the reagent, and the low cost. It was recently found that Oxone® oxidation pretreatment could potentially be substituted with TEMPO-mediated oxidation pretreatment due to existing impediments of TEMPO-mediated oxidation pretreatment such as toxic side products (halogen-containing, co-oxidants), contamination of final products (risk of having trace amounts of reactive radicals in the product), higher cost, and complexity for industrial applications [159,160]. Ruan et al. [157,158] employed Oxone® pretreatment, where 2 g of never-dried cellulose fiber was dispersed in 40 mL deionized water using high-energy ultrasonication. The dispersed cellulose fiber was then incorporated into a round-bottom flask containing 9.1 g Oxone® (2.4 equivalents per anhydroglucose unit) pre-dissolved in 10 mL deionized water. The Oxone® oxidation pretreatment reaction was executed at 80 °C under stirring. After 24 h, the Oxone®-oxidized cellulose fibers were thoroughly washed with deionized water by filtration. It was explored that Oxone® introduced anionic groups on the surface of cellulose fibers through selectively oxidizing the C6 hydroxyl groups of cellulose to carboxylic groups [157,158]. Although Oxone® pretreatment represented a lower carboxyl content (1.073 mmol/gr) of softwood sulfite bleached cellulose fibers than TEMPO-mediated oxidation, the oxidized cellulose fibers could be easily disintegrated. Hence, it means that less mechanical treatment is required to obtain a certain degree of dispersion. It can interestingly be concluded that pretreatment with Oxone® oxidation creates a mechanical fibrillation process that is more efficient and low energy-demanding, however, it undesirably decreases the aspect ratio of the CNFs. The prepared CNFs with this pretreatment had a diameter of about 10 nm, while untreated pulp presented CNFs with a diameter of about 50 nm. The lengths of these individualized CNFs were explored to be a few hundred nm.

5. Commercialization and Challenges

Over the last few years, we have witnessed a worldwide emergence of enterprises, focusing on the production and commercialization of CNFs. Cellulose nanofibers are being produced at different facilities such as industrial, pilot, and lab scales due to its extraordinary properties and pivotal role in the development of a bioeconomy in forest product industries. Table 7 represents a recent list of defined facilities for the production of CNFs worldwide. However, we believe that there are numerous unreported lab scale facilities at universities, research centers, paper mills, and other sites that have not been included in Table 7. Based on the information cited in Table 7, the total capacity to produce CNFs was estimated to be 163 tons/day. Nevertheless, academic and industrial sectors are still clamoring for new pathways toward the scaling up of CNF production and the
A steep rise in demand to develop sustainable materials in various applications supports the need to produce cost-effective CNFs to compete with existing low-cost materials. Diversity in the raw material structures is one of the biggest challenges in reaching a unique method for CNF production. Generally, the chemical composition of the raw materials is species-dependent, and even for a specific plant species, it depends on various factors. Amending the production methods to keep the consistency among the CNF properties from one batch to another result in less rejection in the final product, driving down the final price. The utilization of non-wood species for CNF production is a way to reduce the costs. As mentioned earlier, the production of CNFs mostly required pretreatment approaches and stages. Implementation of new protocols, specifically the one-pot production methods to reduce the chemicals, equipment, and time, would decrease the production costs. Part of the costs can be also decreased either by the development of the existing functionalization methods aiming to increase the charge density of fibers, or by introducing the less aggressive chemicals and more straightforward and green strategies. Inventing new approaches to overcome the clogging issues in the equipment is also a necessity. Finally, finding a major application that needs a large quantity of CNFs is also required for making CNFs industrially worthwhile.

### Table 7. Most recent list of implemented facilities for the commercial production of CNFs globally.

| The Commercial Name of Facility                        | Country Located       | Capacity Tons/Year |
|--------------------------------------------------------|-----------------------|--------------------|
| Kruger Biomaterials Inc.                               | Canada                | 6000               |
| Domtar                                                 | Canada                | 50–75 tons/day     |
| American Process Inc.                                  | USA                   | 100                |
| University of Maine, Process Development Center        | USA                   | 1 ton/day          |
| Turners Falls Paper, Paperlogic                        | USA                   | 2 tons/day         |
| Fibria                                                 | Brazil                | 2 tons/day         |
| Suzano                                                 | Brazil                | 50 kg/day          |
| Borregaard                                             | Norway                | 1000               |
| Norsk Skog                                              | Norway                | 1 ton/day          |
| RISE Bioeconomy                                        | Sweden                | 1 ton/day          |
| Stora Enso Ltd.                                        | Sweden                | Industrial         |
| UPM-Kymmene Ltd.                                       | Sweden/Finland        | Pilot scale        |
| VTT                                                    | Finland               | Semi-pilot         |
| CelloComp                                              | UK                    | 400                |
| Fiber Lean Technologies                                | UK                    | 8000               |
| InTechFibers                                           | France                | 100 kg/day         |
| SAPPI                                                  | Netherland            | 8                  |
| WEIDMANN                                               | Switzerland           | 500                |
| DKS Co., Ltd.                                          | Japan                 | 50                 |
| Daio Paper Corp.                                       | Japan                 | 100                |
| Chuetsu Pulp & Paper Co. Ltd.                          | Japan                 | 100                |
| Nippon Paper Group                                     | Japan                 | 500                |
| Oji Holdings Corporation                               | Japan                 | 40                 |
6. Conclusions and Future Remarks

Cellulose nanofibers (CNFs) are considered as a new bionanomaterial with outstanding intrinsic properties. Due to their properties such as renewability, biocompatibility, and biodegradability, they have abundant applications, more particularly in value-added biomaterials. Although there are numerous operating commercial and pilot plants, CNF and LCNF utilization are still limited, mainly due to high initial investment and production costs, and high mechanical energy demands. This review paper, for the first time, addresses the recent trends and most applied routes of energy-efficient and eco-friendly processes in the production of different CN qualities from low-cost raw materials, with efficient pretreatments and mechanical processes. In order to overcome the available obstacles in making CNFs commercially competitive, new strategies including the use of low-cost raw materials, effective chemical and enzymatic pretreatments, and appropriate selection of final mechanical processes were discussed. Therefore, an update of the most relevant processes and energy-efficient mechanical approaches for low-cost CNFs and LCNFs were practically outlined.

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Nomenclature

| Abbreviation | Description |
|--------------|-------------|
| ACC          | All cellulose composites |
| AQ           | Anthraquinone |
| BKHW         | Bleached kraft hardwood pulp |
| CMC          | Carboxymethyl cellulose |
| COO⁻         | Carboxylate group |
| CaCl₂        | Calcium chloride |
| CNFs         | Cellulose nanofibers |
| CTMP         | Chemithermo mechanical pulp |
| CNCs         | Cellulose nanocrystals |
| CMP          | Chemi-mechanical pulp |
| ChCl         | Choline chloride |
| DPFS         | Date palm fruit stalks |
| DESs         | Deep eutectic solvents |
| DP           | Degree of polymerization |
| NH₄₂HPO₄     | Diammonium hydrogen phosphate |
| PSNFD        | Dried paper sludge nanofibers |
| EFB          | Empty fruit bunches |
| EPFBF        | Empty palm fruit bunch fibers |
| FESEM        | Field emission electron scanning microscopy |
| HPHs         | High pressure homogenizers |
| HPMs         | High pressure microfluidizers |
| HBD          | Hydrogen bond transferor |
| HBR          | Hydrogen bond catcher |
| H₂O₂         | Hydrogen peroxide |
| OH           | Hydroxyl group |
| KWh/Kg       | Kilowatt-hour per kilogram |
LCNFs  Lignocellulosic nanofibers
LNPs  Lignin nanoparticles
LiCl  Lithium chloride
LPMO  Lytic polysaccharide monooxygenase
MKCA6-2J  Masuko super masscolloider
MPa  Megapascal
MFA  Microfibril angle
MFC  Micro-fibrillated cellulose
µmol/g  Micromol per gram
µeq/g  Microequivalent per gram
MCB  Milk container board
MOP  Mixed office paper
MSWP  Municipal solid waste paper
PSNFND  Never-dried paper sludge nanofibers
NSSC  Neutral sulfite semichemical
nm  Nanometer
MCF  Microcrystalline cellulose
OCC  Old corrugated container
PSNF  Paper sludge nanofibers
Ph-GH5  Purified and commercial Endoglucanase
RN  Recycled newsprint
RH  Relative humidity
SiC  Silicon carbide
G’  Storage modulus
VTT  Technical Research Centre of Finland
TEMPO  2,2,6,6-Tetramethylpiperidin-1-yl)oxyl
TSE  Twin screw extruder
VS  Vinyl sulfonate solution
WB  Waste board

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