Perfluorocarbon Emulsion Contrast Agents: A Mini Review

Ryan Holman1 *, Orane Lorton1, Pauline C. Guillemin1, Stéphane Desgranges2, Christiane Contino-Pépin2 and Rares Salomir1,3

1Image Guided Interventions Laboratory (GR-949), Faculty of Medicine, University of Geneva, Geneva, Switzerland, 2University of Avignon, CBSA-IBMM, (UMR5247), Avignon, France, 3Radiology Department, University Hospitals of Geneva, Geneva, Switzerland

Perfluorocarbon emulsions offer a variety of applications in medical imaging. The substances can be useful for most radiological imaging modalities; including, magnetic resonance imaging, ultrasonography, computed tomography, and positron emission tomography. Recently, the substance has gained much interest for theranostics, with both imaging and therapeutic potential. As MRI sequences improve and more widespread access to 19F-MRI coils become available, perfluorocarbon emulsions have great potential for new commercial imaging agents, due to high fluorine content and previous regulatory approval as antihypoxants and blood substitutes. This mini review aims to discuss the chemistry and physics of these contrast agents, in addition to highlighting some of the past, recent, and potential applications.

Keywords: contrast agents, emulsions, perfluorocarbons, fluorine chemistry, radiology

INTRODUCTION

Typical radiological contrast agents are generally iodinated substances for computed tomography (CT) and are gadolinium-based substances for magnetic resonance imaging (MRI). Liquid perfluorocarbon emulsions have been well studied as a diagnostic contrast agent, but has not received regulatory approval for routine clinical use as an intravenous contrast agent by the United States Food and Drug Administration (FDA) or the European Medicines Agency (EMA).

Liquid and gaseous perfluorocarbons have been used in commercial imaging agents. The phase of the perfluorocarbon at physiological conditions is generally dependent on the boiling point, which varies with the molecular weight of the substance; though some substances, like perfluorooctylbromide (PFOB), do not completely adhere to this trend due to intermolecular interactions (Cosco et al., 2015). The optimal perfluorocarbon formula varies with the application. For instance, ultrasonography implements gaseous perfluorocarbon in microbubbles as contrast agents; while, 19F-MRI uses unemulsified gases and high density perfluorocarbon emulsions to increase fluorine signal (Cosco et al., 2015).

Due to the high oxygen solubility, perfluorocarbons were heavily studied as antihypoxants and blood substitutes. Later generations benefited from a reduced side-effect profile, longer storage life, ability to be frozen, ability to be sterilized by autoclave, a more uniform size distribution, and shorter accumulation times in tissue (Vorob’ev, 2009). Many alternative formulas and experimental applications, like radiological contrast agents, have developed alongside. Methods like X-ray CT and MRI allow quantitative concentration measurements and improved tissue contrast (Mattrey et al., 1990; Riess, 2001). Many recent preclinical studies of potential liquid perfluorocarbon emulsions are aimed at theranostic (i.e., therapy and diagnostic) capabilities and alternative applications. The clinical and preclinical studies include drug delivery (Al Rifai et al., 2020),

MINI REVIEW

Published: 10 January 2022
doi: 10.3389/fchem.2021.810029
SYNTHESIS AND PHYSIOCHEMICAL PROPERTIES OF PERFLUOROCARBON CONTRAST AGENTS

Perfluorocarbons can be generated from hydrocarbons by fluorination to substitute the hydrogen atoms with fluorine, with PFOB being of great interest for imaging agents (Riess, 2001). Of primary commercial benefit to the use of PFOB is the capability for large-scale production. The molecule can be derived in a high yield, by a one-step telomerization process through direct bromination of the F-alkyl iodides used in the production of Teflon (Riess, 2001). As perfluorocarbons are immiscible in aqueous solution, these require emulsification for stability (Corvis et al., 2018). The emulsifiers are often phospholipids, poloxamers, or fluorosurfactants. Phospholipids are based on egg lecithin isolated from egg yolk through solvent extraction, and composed largely of phosphatidylcholine (Gobley, 1846; Hensing, 2004). Poloxamer surfactants include Pluronic-F68 used as an emulsifier in Fluosol emulsions and Proxanol-268 used as an emulsifier in Perfloran (Riess, 2001). Poloxamers are made with high polydispersity for other industrial applications, and can be purified for medical grade applications (Riess, 2001). Amphiphilic fluorosurfactants are composed of a fluorinated tail group and a hydrophilic head group, and allow very low interfacial tension (Riess, 2001). The lethal doses (LD₅₀) of some selected poloxamers and fluorosurfactants are Proxanol-268 at 20 g. kg⁻¹, Pluronic-F68 at 9.4 g. kg⁻¹, and F-TAC at 4.5 g. kg⁻¹ in rodents (Vorob’ev, 2009; Maurizis et al., 1994).

The industrial process of large-scale emulsions manufacturing is well developed in pharmaceutics and has long been implemented in the production of parenteral nutrition (Riess, 2001). The emulsion solutions have generally been produced through sonochemical ultrasonic processes which are linked with cavitation, where cavitation nuclei originate from small air bubbles or dust particles in solution, imploding upon excitation to promote further emulsification (Canselier et al., 2002). Cavitation events form when the fluid hydrodynamic pressure becomes lessened to the vapour pressure (Bondy and Solnner, 1935). The vapour pressure of perfluorocarbons in emulsion contrast agents (about 1–3 kPa) are comparable, but slightly lower than water (6.3 kPa) and blood plasma (6.4 kPa) at physiological temperature (Vorob’ev, 2009; Grollman, 1928). These effects will alter the emulsification process, including the droplet diameter and size distribution. Short sonication times tend to generate larger droplets while longer sonication times result in smaller droplet size (Canselier et al., 2002). Cavitation effects can be enhanced with lower ultrasound frequency, lower acoustic pressure, lower medium viscosity, lower medium surface tension, higher energy density, and higher acoustic intensity (Lorimer and Mason, 1987; Canselier et al., 2002). Cavitation implosion effects are also reduced in solvents with higher vapour pressures or at increased temperatures that raise vapour pressure in the fluid-vapour mixture in cavitation sites (Canselier et al., 2002).

Early investigations noted that the incorporation of chlorine or bromine atoms into the perfluorocarbons resulted in faster excretion rates, not predicted based on molecular weights alone (Kabalnov et al., 1992; Riess, 2001). This halogen gives the molecule lipophilic character and enhances clearance rates by allowing the molecule to bind to circulating lipids en route to pulmonary excretion (Long et al., 1972a; Long et al., 1982a; Kabalnov et al., 1992; Weers, 1993; Riess, 2001). Although perfluorocarbons have low hydrocarbon affinity, halogen bonding is observed in some systems, to generate self-assembly of supramolecular and crystalline structures (Fox et al., 2004). In these systems, there is a non-covalent interaction between a halogen atom in a perfluorocarbon molecule that acts as an Lewis acid electron acceptor and an atom that acts as a Lewis base electron donor. Similar to hydrogen bonding effects, the halogen atoms are prone to accept electron density from the free electron pairs in neighboring molecules, as the fluorine atoms have a strong electron withdrawing effect (Fox et al., 2004).

The high gas solubility in the liquid perfluorocarbons can be explained by the nonpolar nature of both molecular species. The perfluorocarbons show low polarity and low polarizability, allowing the molecules to readily dissolve molecular gases like noble gases, oxygen, nitrogen, and carbon dioxide (Riess, 2001; Dias et al., 2004). The low polarizability generates a lipophobic character while the overall nonpolar character leads to hydrophobicity (Riess, 2001; Israelachvili, 2015). Some physical and chemical properties of PFOB are given in Table 1.

COMPUTED TOMOGRAPHY PERFLUOROCARBON EMULSION CONTRAST AGENTS

In computed tomography, an X-ray beam is rotated circumferentially around the patient while measuring the X-ray transmission at each interval (Hasebroock and Serkova, 2009). Iodinated contrast media are common for computed tomography, particularly for angiography, but also for identifying lesions (Hasebroock and Serkova, 2009). These are often substances like Lipiodol® which is derived from poppy seed oil. This medium can be taken up by tumours for cases like hepatocellular carcinoma, where the oil remains longer than in healthy tissue, allowing contrast enhancement on CT images (Rasmussen, 2008). Much initial success for perfluorocarbon contrast enhancement came during studies involving radiopaque brominated perfluorocarbon for radiography, namely PFOB (Long et al., 1982a; Wolf et al., 1994; Hirschl
TABLE 1 | PFOB Physical Properties

| Property                        | Value                  |
|--------------------------------|------------------------|
| Molecular formula               | C₈F₁₇Br                |
| Molecular weight, (g.mol⁻¹)     | 500.0                  |
| Molar Volume, (Å³)              | 432.0                  |
| Boiling point, (°C)             | 143.0                  |
| Melting point, (°C)             | 5.0                    |
| Critical solution temperature   | -20.0                  |
| Solubility in water, (mol.L⁻¹) | 5.6E-6                 |
| Ostwald coefficient for O₂ in PFOB, 308 K | 0.5 | Deschamps et al. (2007) |
| Henry’s constant for O₂ in PFOB, 308 K | 21.6 |
| Diffusion coefficient, m.s⁻¹    | 5.2E-10                |
| Density, (g.m⁻³)               | 1.920.0                |
| Sound velocity, (m.s⁻¹)         | 631.8                  |
| Refractive index (at 298 K)     | 1.3                    |
| Adiabatic compressibility, (kJ.kg⁻¹) | 6.9E-13               |
| Surface tension, (mN.m⁻¹)       | 18.0                   |
| Interfacial tension with water, (mN.m⁻¹) | 48.7 | Astafyeva et al. (2015) |
| Vapour pressure, (kPa at 37°C) | 2.7                    |
| Heat of vaporization, (kJ.mol⁻¹) | 1.3                     |
| LDo₅ in rodents, (g.kg⁻¹)       | 4.8                    |
| Toxic hazard classification by Cramer | High (Class III) |
| DNA binding by OASIS           | AN2                    |
| Genetic toxicity               | Negative               |

*Ostwald coefficient for O₂ in Water at 308 K and 1 atm is about 0.028 (Flettich et al., 2003).

et al., 1996; Riess, 2001; Riess, 2005; Tak and Barraclough, 2018). Perfluorohexylbromide (PFHB) and PFOB, have been long used as CT contrast agents for viewing areas like the bronchia, gastrointestinal tract, and tumours (Patronas et al., 1983).

As the bromine halogen provides radiopacity, intravenous PFOB emulsified with lecithin at doses between 1 and 3 g.kg⁻¹ were previously tested in human studies as a CT contrast agent to image the blood vessels, liver, and spleen (Bruneton et al., 1989). The substance was effective at identifying small metastatic lesions and distinguishing blood vessels from small lesions, compared to non-contrast CT. Tumour lesion enhancement of small metastatic lesions on MRI, CT, and ultrasonography, has been observed after intravenous injection of PFOB emulsions; thought to be attributed to the enhanced permeability and retention effect or macrophage phagocytosis (Mattrey, 1989). Perfluorocarbons have shown utility in diagnostic X-ray radiography for bronchography and alveography in humans (Long et al., 1982b; Vovnikov et al., 1990). CT contrast agent perfluorocarbon emulsions have shown attenuation increases in dogs and pigs of 117 Hounsfield units (HU), 77 HU, and 54 HU in the vasculature (~13–50 HU normal), spleen (~45 HU normal), and liver (~60 HU normal), respectively (Mattrey et al., 1984). These values are very similar to attenuation increases reported with commercial iodinated contrast agents (Amato et al., 2013). It has also been shown as a macrophage-specific CT contrast agent of liver tumours (Vovnikov et al., 1990). Flior-RK has also been reported as an effective contrast agent for X-ray CT attenuation, composed of PFOB and perfluoromethycyclohexylpiperidene (PFMCP), emulsified with poloxamer Proxanol-268, with 50–80 nm diameter, and a circulatory half-life around 24 h (Vorob’ev, 2009; Vorob’ev et al., 1993). Additionally, perflubron injection was used as a contrast agent for lymph node CT imaging in human volunteers with only site injection discomfort as a reported side-effect (Hanna et al., 1994).

MAGNETIC RESONANCE IMAGING PERFLUOROCARBON EMULSION CONTRAST AGENTS

MRI uses radiofrequency (RF) to excite protons and measure the RF emissions as the protons relax to equilibrium. Inside the bore, the protons align in the direction of the scanner’s main, longitudinal, magnetic field. The protons do not align completely, but precess at a resonant frequency around the direction of the net magnetic field (Katti et al., 2011). Then, a RF pulse, with the same frequency as the precessing protons, is absorbed and displaces the spins into the transverse plane. The proton net magnetization vector relaxes towards the main magnetic field, emitting a free induction decay signal that is detected in the RF receiver coil. The amplitude and phase over a range of emitted RF signals is then correlated to the intensity and location to generate the contrast seen in the MRI image (Westbrook, 2016). The signal has a T₁ component in the direction of the main magnetic field and T₂ component transverse to the main magnetic field direction, with the phase lag generating the image contrast. Magnetic resonance contrast agents generally work by reducing the T₁ or T₂ relaxation rates of the tissue in the image, with diseased tissue like tumours having a varying amount of contrast agent than other tissue (Geraldes and
positive contrast agents result in a reduction in $T_1$ relaxation rate to create signal hyperintensity, while negative contrast agents alter the $T_2$ relaxation rate and generate signal hypointensity (Laurent et al., 2009). Perfluorocarbon contrast agents operate through a different mechanism, reducing the local proton density signal to act as a negative contrast agent.

Molecular imaging is a technique to image biomarkers indicative of a disease, by using ligand-mediated agents to target specific cell receptors; for instance to target molecules highly or exclusively expressed in atherosclerotic plaques, ischemic tissues, and tumours (Krafft and Riess, 2021). The liquid perfluorocarbon emulsions have also been studied as a molecular imaging probe. The emulsions can be conjugated with molecular markers specific to certain cell types, including antibodies, peptides, and oligosaccharides; and the accumulation can be visualized with $^{19}$F-MRI (Krafft and Riess, 2021). Other contrast agents like gadolinium chelates, iron oxides, hyperpolarized ions, or fluorodeoxyglucose ($^{18}$F-FDG), can be incorporated into the emulsions to allow contrast enhancement with, $^{19}$F-MRI, $^3$H-MRI, or PET (Wolber et al., 1999; Fabiilli et al., 2013; Amir et al., 2017; Krafft and Riess, 2021). The colloids can also be loaded with drugs and act as theranostics, to both treat the disease and monitor biodistribution (Krafft and Riess, 2021). $^{19}$F-MRI cell tracking with intravenous perfluorocarbon emulsions has been used in humans in a phase I clinical trial for tracking dendritic cell vaccine immunotherapy during treatment of late-stage colorectal cancer (Ahrens et al., 2014), benefitting from being implemented in previous clinical studies for oxygenation after brain trauma and stroke (Darçot et al., 2020).

Imagent GI was a previously FDA-approved unemulsified PFIOB oral negative contrast agent for $^3$H-MRI that has since been discontinued (Brown et al., 1991; Bisset et al., 1996; Geraldes and Laurent, 2009). PFIOB molecules contain no hydrogen atoms and create a proton density weighted image, effectively darkening the bowel loop, allowing better delineation of blood vessels and abdominal organs like the spleen, liver, kidneys, and pancreas (Mattrey et al., 1988; Mattrey, 1989; André et al., 1990; Mattrey et al., 1991). Also, the use of $^{19}$F-MRI with gaseous unemulsified perfluorocarbon as a contrast agent for respiratory disease has completed many early phase clinical trials (Couch et al., 2013; Halaweish et al., 2013; Goralski et al., 2020; Krafft, 2021). Perfluoropropane (PFP) and $^{19}$F-MRI has shown in human trials the ability to distinguish healthy and diseased lungs from patients with COPD, cystic fibrosis, asthma, and emphysema. (Halaweish et al., 2013; Couch et al., 2019; Guterlet et al., 2018) $^{19}$F-MRI of the lungs can provide functional imaging and generally uses a mixture of about 21% oxygen and 79% PFP or sulfur hexafluoride (SF$_6$) (Couch et al., 2013). Healthy volunteers show a homogenous distribution of PFP throughout the lungs, while patients with diseased lungs show incomplete heterogeneous gas distribution. The technique has also shown useful to assess proper lung function from patients that received a lung transplant (Halaweish et al., 2013). The modality would surely be useful in understanding the effects of coronavirus disease 2019 (COVID-19) on lung structure and function, though a specialized MRI receiver coil is required, and to date no studies have reported results; though at least one trial has been initiated (NCT04872309).

**OTHER APPLICATIONS**

Though, radiological studies have indicated the efficacy, many preclinical formulas have yet to undergo costly toxicological safety studies needed for an investigative new drug application, to permit clinical studies on humans. Much of the literature describing toxicological safety studies of perfluorocarbon emulsions has been for blood oxygenation (Spahn, 1999; Leese et al., 2000; Noveck et al., 2000; Hill et al., 2002; Spiess, 2009; Hill, 2019). Perftoran, also known as Perfluorane, is a perfluorocarbon emulsion composed of 10% w:v perfluorodecalin and perfluoromethycyclohexylpiperidine, currently approved as an anti-ischemic and antihypoxant drug in the Russian Federation, Uzbekistan, and Mexico, and was previously approved in many former Soviet states (Khan et al., 2020; Maevsky et al., 2020). Perftoran has been shown to induce vasodilation in patients with vascular disease; including patients with limb ischemia, atherosclerosis, diabetes mellitus, oedema after trauma, and oedema post-surgery (Moroz et al., 2007). These emulsions have also shown to improve preterm birth outcomes during gestosis and preeclampsia, which can exhibit acute damage to the peripheral vasculature, platelet damage, vessel constriction, and organ hyoperfusion; in severe cases, leading to acidosis and organ failure. When used in combination with cytoflavin, improved outcomes have been shown in the treatment of moderate preeclampsia, by increasing vascular perfusion and reducing hypoxia, to prolong pregnancy (Kachalina et al., 2007). Moreover, human studies have shown decreased mortality during severe sepsis by oxygenation and improved microcirculation (Vermolenko et al., 2007). Perftoran has shown improved outcomes when administered intraoperatively during lobectomy of lung cancer patients with severe respiratory disorders (Kligunenko et al., 2007). The developers have indicated the need and utility for re-establishing large-scale industrial production for the treatment of COVID-19 (Maevsky et al., 2020).

Liquid perfluorocarbons are also used extensively in ophthalmological surgery for applications including: giant retinal tears, vitreoretinopathy, and retinal detachment repairs (Kramer et al., 1995; Mikhail et al., 2017). Also, the perfluorocarbon emulsions are under study as potential focused ultrasound adjuvants due to their enhanced absorption of ultrasonic energy and resulting increasing heat generation (Schad and Hynynen, 2010; Zhang et al., 2011; Kopechek et al., 2013; Phillips et al., 2013; Moyer et al., 2015; Desgranges et al., 2019; Lorton et al., 2020). Echogen was a previously FDA approved ultrasonography perfluorocarbon phase-shift emulsion, causing a change from a liquid to gas state when imaged with ultrasonography; composed of C$_5$F$_{12}$ and an albumin surfactant (Lin and Pitt, 2013). Perftoran is effective as an ultrasonography contrast agent in identifying fluid foci liver lesions and for echocardiography (Vakulenko et al., 2021). Flusosol emulsions have been used in early phase clinical
trials as an adjuvant to radiotherapy for high-grade gliomas, and late-stage squamous cell carcinomas in the neck and head (Lustig et al., 1989; Evans et al., 1990).

CLEARANCE, TOXICITY, AND SIDE-EFFECTS

PFOB has been well-studied as an emulsion contrast agent in humans, particularly due to its fast excretion rate (Burgan et al., 1987; Mattrey, 1989); approximately 3 days at 2.7 g. kg⁻¹ (Riess, 2001). Intravenous injection of 1 g. kg⁻¹ 0.1–0.2 μm lecithin-PFOB emulsions in 60 patients gave no detectable toxicity (Mattrey, 1989). Oral administration of unemulsified gastrointestinal PFOB contrast agent at 2–12 ml. kg⁻¹ doses resulted in no toxic symptoms within 3 days in 60 human subjects, with almost all PFOB eliminated within 24 h (Long et al., 1972b). The contrast agent pharmacokinetics can be quantified with MRI, CT, positron emission tomography, gamma counting, high-performance liquid chromatography, and elemental analysis (Pierre and Allen, 2017). Lecithin-PFOB pharmacokinetic studies indicated that intravenous emulsions are opsonized to a large extent by Kupffer cells and splenocytes, resulting in large deposits of PFOB in the liver and spleen within a few minutes of injection (Riess, 2001; Blanco et al., 2015). Here, the emulsions are degraded, then unmetabolized PFOB re-enters the blood stream, binds to plasma lipids, accumulates in the lungs, before being expelled by respiration (Spahn, 1999).

Intravenous lecithin-PFOB at doses between 1 and 3 g. kg⁻¹ were previously tested in humans as a CT contrast agent to image the blood vessels, liver, and spleen (Bruneton et al., 1989). The toxicity was assessed with laboratory tests 2 days before and 7 days following using blood samples, electrolytes, liver function, renal function, proteins, and endocrine factors. The side-effects were mainly asymptomatic and included splenomegaly, and abnormal gamma glutamyl transferase, alkaline phosphatase, and blood platelet levels. Slight lower back pain was observed in some patients thought perhaps to result from venous constriction. These symptoms and influenza-like symptoms are typical of perfluorocarbon emulsions, also seen in liposomal parenteral nutrition formulas, and generally all substances with adsorbent surfaces (Vorob‘ev, 2009). The side-effects have also been linked with a size-dependence, as smaller emulsions are less detectable by the macrophage phagocytosis system (MPS) (Spahn, 1999). A concentration dependence of side-effects has also been linked to the increased release of cytokines that can result in flushing and fever (Flaim, 1994). Early emulsion formulas incorporated Proxanol-F68 emulsifier, later replaced by phospholipids or Proxanol-268, which were designed to avoid the immune system and resulted in improved circulatory half-lives, reduced cytokine response, and reduced side-effects (Vorob‘ev, 2009). Though, not reported to have caused adverse health effects, some long-term accumulation has been suspected from CT imaging of patients whom were administered PFOB for liquid ventilation therapy during severe respiratory distress syndrome (Hagerty et al., 2008; Servaes and Epelman, 2009; Tak and Barraclough, 2018).

The primary constituent in perfluorocarbon contrast agents are a class of chemicals known as perfluoralkyl substances (PFAS). The adverse reports are generally associated with prolonged environmental exposures. Certain PFAS molecules have very long half-lives of 3.5–8 years and have been indicated in many potential adverse health effects (Olsen et al., 2007; Cardenas et al., 2018). The prolonged exposure to PFAS typically occur through contaminated water and food (Cardenas et al., 2018; Fraser et al., 2012; D‘eon and Mabury, 2011). There have been clinical trials and cohort studies of adverse health associated with elevated levels of perfluorooctane sulfonate (PFOS) and perfluorooctanoic acid (PFOA) in blood serum, particularly in pregnant women (Granum et al., 2013; Cardenas et al., 2018; Wikström et al., 2019).

PATH TO CLINICAL TRANSLATION

For FDA investigational new drug applications (iNDA), products must show efficacy and safety through a series of in vitro and in vivo tests. A comprehensive overview of the toxicological testing needed prior to iNDA are given by Andrade et al. (2016) Toxicity studies for contrast agents are generally performed after sufficient image enhancement has been verified, pharmacokinetics are known, and elimination routes have been observed (Pierre and Allen, 2017). The studies are aimed at determining the toxic effects on animals so that the effects can be monitored in human studies and also to determine a limit for no observable adverse effects at higher limits than the desired dose, to create a factor of safety in human studies (Pierre and Allen, 2017). Toxicity studies for MRI contrast agents include local site toxicity, allergies and immunogenicity, genotoxicity, and blood compatibility (Pierre and Allen, 2017).

In vitro test kits offer an affordable alternative to some animal testing and include assays for fetal-embryonic development (EMA/CHIMP/ICH/544278/1998, 2020), gene toxicology assessment, macrophage and neutrophil function assays (CHIMP/167235/2004, 2006), among others. In vitro enzymatic screening, like liver microsomal preparations, are used in most pharmaceuticals to assess metabolism of the substance based on the clearance route (Pierre and Allen, 2017). Perfluorocarbon emulsion contrast agents have been shown to accumulate in the liver and spleen, before being cleared through the lungs. Illustrating safety to these organs would certainly be necessary for translation. Cytochrome P450 enzymes metabolize the majority of drugs in the liver and assay testing the drug reaction can limit adverse effects in patients and establish half-maximal inhibitory concentration (IC₅₀) values (Lynch and Price, 2007). Other in vitro test kits include human colon adenocarcinoma cells (Caco-2) for intestinal permeability, plasma protein binding with ultrafiltration, the Ames test for mutations, micronucleus assay for chromosome damage, and high-throughput screening of hERG channel inhibition for cardiovascular safety effects (Andrade et al., 2016).

Clinical trials for pharmaceutical development generally consist of early phase clinical studies on a small patient cohort
to determine pharmacokinetics, dose, efficacy and assess side-effects (Lipsky and Sharp, 2001). Subsequent late-phase clinical trials are randomized controlled trials on larger groups to further test efficacy, adverse reactions, and long-term safety effects; including post-market trials for rare adverse events (Lipsky and Sharp, 2001). The quickest route for clinical translation of a new perfluorocarbon emulsion contrast agent formula might be using previously approved perfluorocarbon and emulsifier components in the formula, or using an off-label commercial perfluorocarbon emulsion formula. During the COVID-19 pandemic, many medical products have received streamlined clinical testing for the treatment, prevention, and diagnosis of COVID-19 (Avdeev et al., 2019). Remdesivir, for instance, was repurposed from an Ebola virus therapeutic to treat coronavirus disease after being streamlined in the United States, in only 4 months, from new drug application (NDA) submission to emergency use authorization (EUA) (Avdeev et al., 2019). Gaseous perfluorocarbons with 19F-MRI have been previously been tested in early phase clinical trials for respiratory complications (Couch et al., 2013; Halaweish et al., 2013; Goralski et al., 2020; Krafft, 2021) and has great potential for assessing effects of coronavirus disease on pulmonary structure and function; recently being initiated in early phase clinical trials for assessing effects on lungs, vasculature, and the heart from patients with COVID-19 (NCT04872309). Additionally, potential exists with antihypoxant therapies, including blood substitutes and antihypoxants, and the heart from patients with COVID-19 (Avdeev et al., 2019; Moroz et al., 2007; Kligunenko et al., 2007; Kachalina et al., 2007; EU/3/2013/2383, 2021; EU/3/202361, 2021) suggest significant benefit for treating complications associated with severe COVID-19 (Avdeev et al., 2019; Maevsky et al., 2020).

**CONCLUSION**

In this mini review, the physicochemical properties, radiological imaging applications, and previous clinical studies with perfluorocarbon emulsion contrast agents have been discussed. Perfluorocarbons provide useful contrast on CT and MRI and current research with theranostics, molecular imaging, and 19F-MRI have great potential for future commercial medical products. Many alternative therapies, including blood substitutes and antihypoxants, and the heart from patients with COVID-19 (Avdeev et al., 2019; Moroz et al., 2007; Kligunenko et al., 2007; Kachalina et al., 2007; EU/3/2013/2383, 2021; EU/3/202361, 2021) suggest significant benefit for treating complications associated with severe COVID-19 (Avdeev et al., 2019; Maevsky et al., 2020).

**AUTHOR CONTRIBUTIONS**

All authors listed have made a substantial, direct, and intellectual contribution to the work and approved it for publication.

**FUNDING**

This project has received funding from the European Union’s Horizon 2020 research and innovation programme under the Skłodowska-Curie grant agreement No 813766.
Nonthermal Ablation in Mice Brain. *The J. Acoust. Soc. America* 144 (3), 1889. doi:10.1121/1.5068282

Phillips, L. C., Puett, C., Sheeran, P. S., Dayton, P. A., Wilson Miller, G., and Matsunaga, T. O. (2013). Phase-shift Perfluorocarbon Agents Enhance High Intensity Focused Ultrasound Thermal Delivery with Reduced Near-Field Heating. *The J. Acoust. Soc. America* 134 (2), 1473–1482. doi:10.1121/1.4812866

Pierre, V. C., and Allen, M. J. (2017). Contrast Agents for MRI: Experimental Methods. London, UK: Royal Society of Chemistry.

Rasmussen, F. (2008). “Contrast Media, Iodinated, Oily,” in Encyclopedia of Diagnostic Imaging. Editor A. L. Baert (Berlin, Heidelberg: Springer Berlin Heidelberg), 501–503.

Rettich, T. R., Battino, R., and Wilhelm, E. (2000). Solubility of Gases in Liquids. 22. Rettich, T. R., Battino, R., and Wilhelm, E. (2000). Solubility of Gases in Liquids. 22. Rettich, T. R., Battino, R., and Wilhelm, E. (2000). Solubility of Gases in Liquids. 22. Rettich, T. R., Battino, R., and Wilhelm, E. (2000). Solubility of Gases in Liquids. 22.

Riemann, J. (2005). Understanding the Fundamentals of Perfluorocarbon Emulsions as Intravenous Delivery Media for Hyperpolarized Xenon. *Magn. Reson. Med.* 41 (3), 442–449. doi:10.1002/mrd.10598

Riess, J. G. (2001). Oxygen Carriers (“Blood Substitutes”) Raison d’Etre, Chemistry, and Some Physiology But is ein ganz besonderer Saft. *Chem. Rev.* 101 (9), 2797–2920. doi:10.1021/cr970143c

Riess, J. G. (2005). Understanding the Fundamentals of Perfluorocarbons and Perfluorocarbon Emulsions Relevant to In Vivo Oxygen Delivery. *Artif. Cell Blood Substitutes, Biotechnol.* 33 (1), 47–63. doi:10.1081/bio-200046659

Schad, K. C., and Hynynen, K. (2010). In Vitro characterization of Perfluorocarbon Droplets for Focused Ultrasound Therapy. *Phys. Med. Biol.* 55 (17), 4933–4947. doi:10.1088/0031-9155/55/17/004

Servaes, S., and Epelman, M. (2009). Perfluorob Residua: 12 Years Following Therapy. *Pediatr. Radiol.* 39 (4), 393–395. doi:10.1007/s00247-008-1139-8

Spahn, D. R. (1999). Blood Substitutes Artificial Oxygen Carriers: Perfluorocarbon Emulsions. *Crit. Care* 3 (5), R93. doi:10.1186/cc364

Spiess, B. D. (2009). Perfluorocarbon as a Promising Technology: a Review of Tissue and Vascular Gas Dynamics. *J. Appl. Physiol.* 106 (4), 1444–1452. doi:10.1152/japplphysiol.09995.2008

Tak, S., and Barradough, M. (2018). Case Report: ‘Pseudo-Calculations’: Detection of Perfluorocarbon Residue on a Computed Tomography Scan 15 Years after Liquid Ventilation Therapy at 3 Months of Age. *BMJ Case Rep.* 2018, bcr2017233958. doi:10.1136/bcr-2017-223958

Vakulenko, I. P., Khatsko, V. V., Kuzmenko, O. Y., Voytyuk, V. N., Fominov, V. M., Polulyach-Chornovil, I. F., et al. (2021). Radial Diagnostics of Liquid Focal Liver Formations with Their Identification of a Communications with the Intrahepatic Biliary Ducts. *Khark. Surg. Sci.* 2021 (1), 40–45. doi:10.37699/2308-7005.1.2021.08

Vorob’ev, S. I. (2009). First- and Second-Generation Perfluorocarbon Emulsions. *Pharm. Chem. J.* 43 (4), 209–218. Vorob’ev, S., Ivanitskii, G., and Makarov, K. (1993). *Perfluorocarbon Emulsions.* Pushchino: Department of Scientific and Technical Information, Ross Acad Sci. Voronkov, T., Nikulova, I., Suzuki, A., Higashino, H., Petkova, M., and Mega, N. (1990). A Review on the Properties and Applications of Perfluorocarbon Emulsions. *Acta Med. Kinki Univ.* 15 (1), 1–20.

Westbrook, C. (2016). “MRI at a Glance,” in *At a Glance*. 3rd ed. (Wiley-Blackwell).

Wikström, S., Lindh, C. H., Shu, H., and Bornehag, C. G. (2019). Early Pregnancy Serum Levels of Perfluoroalkyl Substances and Risk of Preeclampsia in Swedish Women. *Sci. Rep.* 9 (1), 9179. doi:10.1038/s41598-019-45483-7

Wolber, J., Rowland, E. J., Leach, M. O., and Bifone, A. (1999). Perfluorocarbon Emulsions as Intravenous Delivery Media for Hyperpolarized Xenon. *Magn. Reson. Med.* 41 (3), 442–449. doi:10.1002/mrd.10598

Wolf, G., Rogowska, J., Gazelle, G., and Halpern, E. (1994). Methods for Quantitative CT Lymphography. *Lymphology* 27, 261–264.

Yermolenko, S. V., Shapovalova, N. V., and Lavrentyev, A. A. (2007). Experience in Using Perfluorane for Severe Sepsis. *Gen. Resusc* III, 67–70.

Zhang, M., Fabilli, M. L., Haworth, K. J., Padilla, F., Swanson, S. D., Kripfgans, O. D., et al. (2011). Acoustic Droplet Vaporization for Enhancement of Thermal Ablation by High Intensity Focused Ultrasound. *Acad. Radiol.* 18 (9), 1123–1132. doi:10.1016/j.acra.2011.04.012

Conflict of Interest: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

Publisher’s Note: All claims expressed in this article are solely those of the authors and do not necessarily represent those of their affiliated organizations, or those of the publisher, the editors and the reviewers. Any product that may be evaluated in this article, or claim that may be made by its manufacturer, is not guaranteed or endorsed by the publisher.

Copyright © 2022 Holman, Lorton, Guillemin, Desgranges, Contino-Pépin and Salomir. This is an open-access article distributed under the terms of the Creative Commons Attribution License (CC BY). The use, distribution or reproduction in other forums is permitted, provided the original author(s) and the copyright owner(s) are credited and that the original publication in this journal is cited, in accordance with accepted academic practice. No use, distribution or reproduction is permitted which does not comply with these terms.