Alterations in the Plasma and Red Blood Cell Properties in Patients with Varicose Vein: A Pilot Study

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The varicose vein results from the inefficient functioning of the valves in the lower limb veins, making the blood flow slow down and leading to blood stasis and hypoxia. This type of vein dysfunction might be a result of the development of oxidative stress. We compared oxidative stress markers in the plasma and erythrocytes obtained from peripheral veins and varicose veins in the same patients (glutathione, nonenzymatic antioxidant capacity (NEAC), catalase (CAT) and acetylcholinesterase (AChE) activity, thiols, thiobarbituric acid-reactive substance (TBARS), and protein carbonyls). We found a decrease in NEAC in the plasma obtained from the varicose veins compared to the peripheral veins. We detected a decrease in thiols in the plasma, hemolysate, and plasma membranes and increase in protein carbonyl compounds and TBARS levels in the varicose veins. These changes were accompanied by a decrease in CAT and AChE activity. For the first time, our results show changes in the plasma, erythrocyte membrane, and hemolysate protein properties in varicose vein blood in contrast to the plasma and erythrocytes in peripheral vein blood from the same patients. The increased oxidative stress accompanying varicose vein disease might result from the local inefficiency of the antioxidant defense system.

1. Introduction

For a long time, chronic venous disease (CVD) and its main clinical phenomenon—varicose veins (VV)—were often neglected and considered an aesthetic problem. Varicose veins are dilated, tortuous veins that make blood flow indolent and stagnant, causing a higher risk of thromboembolic disease. The presence of varicose veins is the most common clinical symptom of chronic venous insufficiency or CVD affecting adult patients. Many genetic, hormonal, and environmental factors trigger the development of venous system diseases, but age and pregnancy have been found to perform the most prominent role in developing varicose veins [1]. The hallmark of varicose veins is the insufficiency of the venous valves; however, the exact cause and molecular mechanisms leading to such a primary dysfunction have not yet been well identified [2]. The main destructive factor is venous hypertension, which may lead to the remodeling of vein walls and vein valves due to an increase in venous pressure, which leads to structural and functional changes in the vein wall. Various studies have hypothesized that the cause of CVD is the trapping of leukocytes owing to the defective valves and oxidative stress in the veins [3–5]. In addition, hypoxia performs a significant role in the physiopathology of varicose veins [6]. In hypoxia, there is a low-oxygen concentration and a high carbon dioxide concentration, which lead to a decline in pH and, consequently, a rise of free iron ions from transferrin. Iron collection causes enhanced catalysis of free radical reactions that produce other reactive oxygen species (ROS) in the activated inflammatory cells present in venous blood, as well as those present in the wall of varicose veins. This leads to the inflammation of the iron-controlled CVD of the lower limbs [3, 7].
It is postulated that the dysfunction of the vascular endothelium might be the result of an excessive generation of ROS, depletion of antioxidant defenses, and the development of oxidative stress [8]. Eventually, the oxidative stress progresses to CVD, creating a vicious circle. Elevated levels of ROS in varicose veins may result in the plasma and blood cell component oxidation, as evidenced by the significant increase in the presence of ROS metabolites in the plasma derived from varicose veins as compared to the levels of ROS in the plasma obtained from healthy donors [5, 9–11]. Other studies have reported changes in the rheological properties of erythrocytes in varicose veins and, therefore, ROS may lead to the oxidation of proteins and lipids within the erythrocytes’ structure [12, 13]. Moreover, changes in the fluidity of membrane lipids and conformational changes in the membrane proteins of erythrocytes present in varicose veins have been observed [13]. Despite numerous studies, the mechanisms associated with varicose vein pathogenesis are still not clear.

Our preliminary study aimed to identify the difference in the oxidative stress markers in the plasma, hemolysate proteins, and erythrocyte membrane samples obtained from varicose vein blood and normal peripheral vein blood (control) in the same patients with CVD.

2. Materials and Methods

2.1. Chemicals. The following chemicals were purchased from Sigma Chemical Co. (St. Louis, MO, USA): 4-iodoacetamide-TEMPO (iodoacetamide spin label; ISL), 4-Amino-TEMPO (temppamine), 2, 4, 6-tripyridyl-s-triazine (TPTZ), o-phthalaldehyde (OPA), xylenol orange, 5-dithiobis-2-nitrobenzoic acid (DTNB), acetylthiocholine (TPTZ), o-phthalaldehyde (OPA), xylenol orange, 5-Amino-TEMPO (tempamine), 2, 4, 6-tripyridyl-s-triazine (TPTZ), o-phthalaldehyde (OPA), xylenol orange, 5-dithiobis-2-nitrobenzoic acid (DTNB), acetylthiocholine iodide, 2, 4-dinitrophenylhydrazine (DNPH), and 2, 2′-dithiodipyridine. Bis-(2, 2, 5, 5-tetramethyl-3-imidazoline-1-oxyl-4-yl) disulfide biradical (RSSR•) was obtained from Sigma Chemical Co. (St. Louis, MO, USA): 4-iodoacetic acid, 4-(iodoacetyl)aminobenzene, 2, 4-dinitrophenylhydrazine (DNPH), and 2, 2′-dithiodipyridine. Bis-(2, 2, 5, 5-tetramethyl-3-imidazoline-1-oxyl-4-yl) disulfide biradical (RSSR•) was obtained from Sigma Chemical Co. (St. Louis, MO, USA). All other reagents of analytical purity were obtained from POCH S.A. (Gliwice, Poland).

2.2. Subjects. Blood samples were collected from eight patients (6 male, 2 female; mean age 54.4 ± 10.6 years; mean BMI 27.7 ± 3.0 kg/m²) with varicose veins classified as class CEAP. The highly selected study group of patients did not suffer from any other chronic or acute disease, including inflammatory etiology diseases, and did not use either phlebotropics or drugs that could affect the blood coagulation process. In addition, in the group of patients, the INR (International Normalized Ratio) (1.03 ± 0.13), sodium (134.4 ± 1.0 mmol/L), and potassium (4.55 ± 0.42 mmol/L) levels in the blood were tested.

Blood samples were taken from a varicose vein and from the antecubital vein from the same patients and transferred into heparinized test tubes. Blood samples were centrifuged to separate the plasma (3000×g for 10 minutes). The erythrocytes were washed three times by centrifugation with phosphate-buffered saline (5 mmol/L, pH 7.4). The procedure was performed at a temperature of 4°C.

All experiments were conducted in accordance with the principles of the Helsinki Declaration and were in accordance with the ethical principles set out in the Belmont Report: Ethical Principles and Guidelines for the Protection of Human Subjects of Research. The test procedure was accepted by the Bioethics Committee of the Medical University of Lodz. All patients agreed to participate in the study by signing an informed consent document.

2.3. Erythrocyte Membrane and Hemolysate Preparation and Protein Determination. The erythrocyte membranes were prepared using the method described by Dodge et al. [14]. The concentration of the plasma and membrane proteins was determined spectrophotometrically with the Folin and Ciocalteu’s phenol reagent according to the method outlined by Lowry et al. [15]. The method was based on the reaction of Cu²⁺, produced by the oxidation of peptide bonds, with Folin–Ciocalteu reagent. The products of this reaction are optically active with a maximum absorption at 750 nm. The amount of protein in each sample was estimated using a calibration curve plotted by taking different concentrations of bovine albumin as a standard.

The hemolysate was prepared from the erythrocytes after they were lysed with cold water at a ratio 1:1.5 and vortexed for 10 min. Hemolysate was centrifuged at 16000×g for 10 minutes, for the separation of erythrocyte membranes. In the obtained hemolysate, the concentration of hemoglobin (Hb) was estimated using Drabkin’s method based on the oxidation of hemoglobin to methemoglobin in the presence of alkaline potassium ferricyanide [16]. Methemoglobin reacts with potassium cyanide to optically active (540 mN) stable cyannmethemoglobin. The molar absorption coefficient of hemoglobin was used to calculate the protein concentration in the samples (ε = 44 mmol⁻¹.L.cm⁻¹).

2.4. Spin Labeling of Hemolysate Proteins. We have used electron paramagnetic resonance (EPR) spin label, ISL, specific to proteins, which binds to protein-SH residues. The conformational changes of hemoglobin in the hemolysate were estimated by measuring the relative rotational correlation time (τc) for ISL attached to Hb as described in an earlier report [17]. The hemolysate was labeled with ISL, dissolved in an ethanol solution (50:1) and incubated for one hour at room temperature. Unbound spin label was eliminated by dialysis in 20 mmol/L phosphate buffer, pH = 7.4, for 24 hours at 4°C. The τc was determined from the EPR spectra according to the formula described in an earlier report [18]:

\[ \tau_c = \frac{1}{2} k w_0 \left( \frac{h_0}{h_{11}} + \frac{h_0}{h_{-1} - 2} \right), \]

where \( h_{11} \) is the low-field line height, \( h_0 \) is the mid-field line height, \( h_{-1} \) is the high-field line height, \( w_0 \) is the mid-field line width, and \( k \) is a constant equal to 1.19×10⁻⁹ s.
The EPR spectra were recorded on a Bruker ESP 300E spectrometer at room temperature (22 ± 2°C), operating at a microwave frequency of 9.73 GHz. The instrumental settings were as follows: the microwave power was set at 10 mW, the center field was set at 3480 G with a range of 80 G, and the modulation frequency and modulation amplitude were set at 100 kHz and 1.01 G, respectively.

2.5. Plasma Viscosity Determination. The plasma viscosity was measured using 4-amino-TEMPO as described by Morse [19]. The spin label in the ethanol solution was added to the plasma and remained unbound (final concentration of 4-amino-TEMPO 1 mmol/L) for 0.5 h at room temperature. The relative time of the rotational correlation ($\tau_r$) was calculated from the EPR spectra on the basis of the following equation [20]:

$$\tau_r = kw_0 \left( \sqrt{\frac{H_0}{H_0 - 1}} \right),$$

where $k$ is a constant equal to $6.5 \times 10^{-10}$ s.

The plasma viscosity was calculated according to the following formula:

$$\eta = \frac{\tau_r(\text{plasma})}{\tau_r(\text{H}_2\text{O})} \eta_{\text{H}_2\text{O}},$$

where $\tau_r(\text{plasma})$ is the rotational correlation time for 4-amino-TEMPO in the plasma, $\tau_r(\text{H}_2\text{O})$ is the rotational correlation time for 4-amino-TEMPO in water, and $\eta_{\text{H}_2\text{O}}$ is the viscosity of water equal to 1 cP.

2.6. Catalase Activity. Catalase (CAT) activity in the hemolysate and the plasma was estimated using hydrogen peroxide ($\text{H}_2\text{O}_2$) as the substrate [21]. The results were reported in units of CAT activity per milligram of Hb or plasma protein (U/mg protein/min), where 1 U of CAT is defined as the amount of enzyme needed to decompose 1 µmol of $\text{H}_2\text{O}_2$ per minute.

2.7. Acetylcholinesterase Activity Determination. The AChE activity in the erythrocyte membranes was measured using the Ellman spectrophotometric method [22]. Data are expressed as µmol/mg protein/minute.

2.8. Determination of Total Nonenzymatic Antioxidant Capacity. The NEAC of the plasma and hemolysate was determined using the ferric-tripyridyltriazine (Fe(III)-TPTZ) complex. In this method, ([Fe(III)-TPTZ]) complex is reduced by cellular antioxidants to a ferrous [Fe-(II)] complex [23].

NEAC was also estimated by the reduction of 2, 2'-diphenyl-1-picrylhydrazyl (DPPH) [24]. A calibration curve was prepared for both methods using different concentrations (0–1000 µmol/L) of Trolox. The results were expressed as nmol of Trolox equivalents per milligram protein or Hb.

2.9. Determination of Glutathione Concentration. The concentration of reduced GSH in the plasma and hemolysate was estimated using the fluorescent method. The reaction of GSH with o-phthalaldehyde led to a fluorescent product. The fluorescence was measured at 365 nm and 430 nm excitation [25]. GSH concentration was read from the calibration curve prepared from various concentrations of GSH and expressed as nmol/mg protein or Hb.

2.10. Determination of Thiol Content. The level of thiols in the membrane and plasma proteins was estimated using Ellman’s reagent ((5, 5'-dithiobis (2-nitrobenzoic acid); DTNB) [26]. The absorbances were measured at a wavelength of 412 nm.

The thiols level in the hemolysate was determined using the Egwim and Gruber method with 4, 4'-dithiodipyridine [27]. The 2-thiopyridone formed in the reaction of thiols with 4, 4'-dithiodipyridine was measured at 324 nm.

For both methods, a calibration curve was prepared from different concentrations of reduced GSH. Data are expressed as nmol thiol/mg protein or Hb.

2.11. Determination of Protein Carbonyl Compound Concentration. The protein carbonyls content in the plasma and membrane was determined with DNPH [28]. The concentration of protein carbonyl compounds was calculated using the millimolar absorption coefficient ($\epsilon = 22 \text{ mmol}^{-1}\cdot\text{L}^{-1}\cdot\text{cm}^{-1}$) and expressed as nmol/mg protein.

2.12. Determination of Thiobarbituric Acid-Reactive Substances. The lipid peroxidation in the plasma was measured using a TBARS assay [29] with modifications by Rice-Evans et al. [30]. The TBARS levels were calculated using the malondialdehyde (MDA) absorption coefficient ($\epsilon = 156 \text{ mmol}^{-1}\cdot\text{L}^{-1}\cdot\text{cm}^{-1}$) and were expressed as micromoles per milligram protein (µmol/mg protein).

2.13. Determination of Peroxides. The peroxides in the plasma membrane were estimated using xylene orange. In the presence of peroxides, Fe (II) is rapidly oxidized to Fe (III) [31]. The concentration of peroxides was calculated from the calibration curve obtained for different concentrations of $\text{H}_2\text{O}_2$ as the standard. Data are expressed as mmol/mg Hb.

2.14. Statistical Analysis. The obtained results were subjected to a statistical analysis using the Shapiro-Wilk test to determine the regularity of the distribution. The homogeneity of the variance was checked using the Brown-Forsythe test. The results fulfilled the conditions of normality.

The differences were determined with Student's $t$-test paired data. All data were presented as mean ± SD and median and interquartile range (IQR: from lower quartile Q1 to upper quartile Q3).

The statistical analysis was performed using Statistica v. 13.0.
3. Results

In this study, we determined the properties of the plasma, erythrocyte membranes, and hemolysate obtained from varicose vein blood and normal peripheral vein blood (control) in the same patients. Table 1 shows the morphological characteristics of blood taken from the peripheral and varicose veins of the studied patients. The viscosity of the plasma from both sources was determined using a tempamine spin label. There was no difference in the viscosity between the varicose vein and normal vein plasma (Table 2). As shown in Figure 1(a), in the plasma from the varicose vein, the level of free thiols was significantly lower than in the peripheral vein plasma ($p < 0.05$). On the other hand, we did not find any difference in GSH concentration in the varicose and peripheral vein plasma samples (Table 2). Furthermore, in the varicose vein plasma, a significantly higher level of TBARS and protein carbonyl compounds was observed in comparison to the peripheral vein plasma (Figures 1(b) and 1(c)).

To determine the total NEAC of the plasma, two independent methods were used. The changes in the NEAC measured with DPPH showed a significant decrease of NEAC in the plasma of a varicose vein in comparison to the plasma from a peripheral vein ($p < 0.05$) (Figure 2). The results obtained from the DPPH were similar to those observed using the ferric reducing ability of the plasma FRAP method (Figures 2(a) and 2(b)). Moreover, in the varicose vein plasma, the CAT activity was significantly lower than in the peripheral vein plasma (Figure 2(c)).

The EPR spectra of the protein-specific spin labels of maleimide and iodoacetamide used accurately reflect the Hb conformation in peripheral and varicose veins hemolysate. The EPR spectra of spin-labeled Hb with ISL in hemolysate were similar to the spectra of spin-labeled purified hemoglobin. The mobility of ISL attached to Hb from varicose blood was significantly higher compared to the mobility of spin-labeled Hb from a peripheral vein ($p < 0.05$) (Figure 3(a)). This result showed alterations in the structure of this protein in varicose veins compared to peripheral vein blood samples. Moreover, we detected a decrease in the total thiols in the hemolysate from varicose vein in comparison to the peripheral vein hemolysates ($p < 0.05$) (Figure 3(b)). Despite the reduced levels of thiols, the concentration of GSH in erythrocytes from varicose vein and peripheral vein blood did not show any differences (Table 2). Similarly, no significant differences in the total NEAC in the hemolysate from varicose vein and peripheral vein blood were observed (Table 2). Further, the antioxidant enzyme CAT activity was determined in the erythrocyte hemolysate. We observed a significantly lower activity in CAT obtained from varicose vein hemolysate in comparison to the activity determined in peripheral vein hemolysates ($p < 0.05$) (Figure 3(c)).

We also evaluated the oxidative stress parameters in the erythrocyte membranes derived from varicose vein and peripheral vein blood. In the case of erythrocyte plasma membranes, a significant decrease in total thiols from varicose vein blood was found in comparison to peripheral vein blood ($p < 0.05$) (Figure 4(a)). Moreover, we observed a notable increase in the protein carbonyl concentration in erythrocyte membrane proteins from varicose vein blood in comparison to peripheral vein blood ($p < 0.05$) (Figure 4(b)). Furthermore, our results reported a reduced AChE activity in varicose vein erythrocyte membranes compared with those of peripheral veins ($p < 0.05$) (Figure 4(c)). However, the concentration of peroxides was found to be the same in erythrocyte membrane proteins derived from both varicose vein and peripheral vein blood (Table 2).

4. Discussion

Varicose veins are enlarged, twisted veins with a diameter of more than 3 mm, and they are the most common lower limb vessel syndrome in humans. Varicose veins result from various factors, such as insufficiency of the vein valves, vessel occlusion, or venous hypertension associated with blood reflux and muscle pump dysfunction [32]. These factors lead to blood stagnation, hypoxia, and the development of chronic inflammation. The hypoxia inducible factor (HIF) and various transcription factors released under hypoxic conditions upregulate the expression of several genes involved in promoting survival under low-oxygen conditions. Hypoxia induces an expression of glycolysis pathway enzyme to generate ATP in an oxygen-independent manner and promoting an expression of vascular endothelial growth factor required for angiogenesis. According to a study, HIF-1a, HIF-2a, and hypoxia perform a crucial role in the pathological process of CVD [33]. More recent studies have shown that the HIF pathway may lead to pathophysiological changes in the varicose vein wall, and hypoxia may contribute to the pathogenesis of VV [34]. However, hypoxia also leads to the increased generation of ROS and, thus, oxidative stress [35, 36]. There are various sources of ROS production, for example, cytoplasmic membranes, endoplasmic reticulum, lysosomes, mitochondria, or peroxisomes [37, 38]. The most efficient route for ROS production in the mitochondrial respiratory chain is during electron transport, with 11 sites producing superoxide and/or hydrogen peroxide [39]. Supposedly, 1% to 3% of oxygen running through the mitochondria is reduced to $O_2^*$ [40].

In fact, the generation of hydroxyl and alkyl radicals in porcine pulmonary arteries under hypoxic conditions has been demonstrated using spin trapping EPR spectroscopy [41].

Another significant manifestation of CVD is the inflammatory response, due to the interaction of neutrophils and other phagocytic cells, causing the release of various inflammatory cytokines, such as tumor necrosis factors, interleukins (ILs), lymphokines, monokines, and ROS [42]. Activated neutrophils release powerful oxidants that can modify extracellular macromolecules and induce oxidative changes in neighboring cells [43]. Additionally, neutrophils can be activated in the plasma of varicose vein blood and serve as an essential contributing factor in the pathogenesis of primary venous dysfunction [44]. Blood drawn from varicose veins appears to have notably increased concentrations of proinflammatory cytokines IL-6, IL-8, and...
Table 1: Clinical hematology parameters in patients with venous disease.

| Parameter          | Peripheral vein | Varicose vein | Statistical significance |
|--------------------|-----------------|---------------|-------------------------|
| WBC (x10^3/µL)    | 7.30 ± 1.83     | 7.97 ± 2.73   | 0.669                   |
| RBC (x10^6/µL)    | 4.71 ± 0.56     | 4.69 ± 0.59   | 0.985                   |
| HGB (g/dL)        | 14.02 ± 1.34    | 14.16 ± 1.88  | 0.673                   |
| HCT (%)           | 40.62 ± 3.32    | 41.27 ± 4.67  | 0.612                   |
| MCV (fL)          | 86.67 ± 4.50    | 88.21 ± 4.92  | 0.155                   |
| MCH (pg)          | 29.98 ± 1.43    | 30.21 ± 1.88  | 0.170                   |
| MCHC (g/dL)       | 34.48 ± 1.06    | 34.26 ± 1.00  | 0.849                   |
| PLT (x10^3/µL)    | 222.70 ± 77.17  | 191.29 ± 46.28| 0.473                   |
| RDW-SD (fL)       | 41.01 ± 3.31    | 41.79 ± 2.80  | 0.828                   |
| RDW-CV (%)        | 13.18 ± 0.85    | 13.24 ± 0.96  | 0.898                   |
| MPV (fL)          | 10.21 ± 0.70    | 10.54 ± 1.03  | 0.062                   |
| NEUT (%)          | 58.13 ± 11.17   | 52.54 ± 17.49 | 0.354                   |
| NETU (x10^3/µL)   | 4.30 ± 1.49     | 4.50 ± 2.79   | 0.870                   |
| LYMPH (%)         | 30.36 ± 9.76    | 33.70 ± 14.44 | 0.605                   |
| LYMPH (x10^3/µL)  | 2.18 ± 0.83     | 2.37 ± 0.51   | 0.887                   |
| MONOC (%)         | 8.51 ± 2.06     | 9.74 ± 2.68   | 0.052                   |
| MONOC (x10^3/µL)  | 0.61 ± 0.20     | 0.73 ± 0.21   | 0.944                   |
| EOS (%)           | 2.56 ± 1.65     | 3.56 ± 2.83   | 0.080                   |
| EOS (x 10^3/µL)   | 0.18 ± 0.10     | 0.22 ± 0.12   | 0.096                   |
| BASO (%)          | 0.44 ± 0.18     | 0.46 ± 0.21   | 0.289                   |
| BASO (x10^4/µL)   | 0.31 ± 0.14     | 0.33 ± 0.13   | 0.356                   |

Data are shown as mean ± standard deviation range, n = 8, and significance level for Student’s t-test. Abbreviations: BASO: basophils; EOS: eosinophils; HCT: hematocrit; HGB: hemoglobin concentration; LYMPH: lymphocytes; MCH: mean corpuscular hemoglobin; MCHC: mean corpuscular hemoglobin concentration; MCV: mean corpuscular volume; MONOC: monocytes; MPV: mean platelet volume; NEUT: neutrophils; PDW: platelet distribution width; PLT: platelets; RBC: red blood cells; RDW-CV: red cell distribution width, coefficient of variation; RDW-SD: red cell distribution width, standard deviation; WBC: white blood cells.

Table 2: Determined parameters of the plasma, hemolysate, and erythrocytes membrane isolated from blood obtained from peripheral vein and varicose vein blood.

| Plasma                | Peripheral vein | Varicose vein | Statistical significance |
|-----------------------|-----------------|---------------|-------------------------|
| Viscosity (ƞ)         | 1.85 ± 0.04     | 1.78 ± 0.07   | n.s.                    |
| Reduced GSH (mmol/mg protein) | 0.0709 ± 0.007  | 0.0692 ± 0.009| n.s.                    |
| Hemolysate            |                 |               |                         |
| Reduced GSH (mmol/mg Hb) | 0.092 ± 0.028  | 0.086 ± 0.025 | n.s.                    |
| NEAC (FRAP) (mmol/mg Hb) | 61.95 ± 4.35  | 65.27 ± 5.08  | n.s.                    |
| Erythrocytes membrane |                 |               |                         |
| Peroxides (mmol/mg protein) | 7.17 ± 0.70  | 6.98 ± 1.56   | n.s.                    |

Data are shown as mean ± standard deviation range as well as median and quartile 1 and quartile 3 (Me; Q1; Q3), n = 8.

Figure 1: The levels of free thiols (a), TBARS (b), and protein carbonyl compounds (c) in the plasma obtained from peripheral and varicose veins. Data are shown as mean ± range of standard deviation, n = 8. *p < 0.05 is the significant difference.
monocyte chemoattractant protein-1 (MCP-1) in comparison to the blood drawn from the same patient’s antecubital vein [45]. Therefore, inflammation in the veins of the lower extremities is more likely than in the veins of the upper extremities. Additionally, it was shown that inflammatory cells, especially leukocytes, perform a key role in both the aging and varicose processes [46]. In addition to being present in vessel walls, ROS can also be found in the plasma due to neutrophil activation.

In our study, we observed lipid and protein peroxidation products, indicated by higher levels of TBARS and protein carbonyl compounds in the plasma of varicose vein patients. The increase in TBARS and protein carbonyl levels was in agreement with the results obtained by Condezo-Hoyos and colleagues, who compared patients with a varicose vein with a group of healthy volunteers [9]. Oxidative damage to proteins and lipids may be associated with a decrease in the level of thiols, as well as a decrease in the level of NEAC in

**Figure 2:** Total nonenzymatic antioxidant capacity in the plasma, determined by (a) DPPH and (b) FRAP methods, and catalase activity (c) in the plasma of the blood samples obtained from peripheral and varicose veins. Data are presented as mean ± range of standard deviation, $n = 8$. *$p < 0.05$ is the significant difference.

**Figure 3:** Mobility of ISL labeled hemoglobin (a), the levels of total thiols (b), and catalase activity (c) in the hemolysate obtained from a peripheral vein and a varicose vein. Data are shown as mean ± range of standard deviation, $n = 8$. *$p < 0.05$ is the significant difference.

**Figure 4:** The levels of free thiols (a), protein carbonyl compounds (b), and acetylcholinesterase activity (c) in the erythrocyte membrane obtained from a peripheral vein and a varicose vein. Data are shown as mean ± range of standard deviation, $n = 8$. *$p < 0.05$ is the significant difference.
the plasma of varicose veins, which was determined using two independent methods. Additionally, we also found a decrease in CAT activity. It was reported that CAT activity can be inhibited by hydroxyl radicals, superoxide, and H$_2$O$_2$ but not by organic peroxides [47, 48].

Our previous report showed changes in the erythrocytes from varicose veins in membrane proteins and lipid membrane fluidity [13]. The present study aims to find the reason for these alterations in the internal viscosity of erythrocytes, as well as antioxidant status in internal fluids and membranes. Using covalently bound ISL, we observed an increase in the $\tau_c$, reflecting the immobilization of spin label residue. Since the main protein present inside the erythrocytes is Hb, the increased correlation time could suggest conformational state changes in this protein. Moreover, ISL has been reported to bind to $\beta$ globin chain ($\beta$Cys93) in Hb [49]. It was reported that $\beta$Cys93 cysteine may influence the redox state of heme iron. Additionally, the oxidative modification of cysteine residue by nitric oxide (NO) leads to the formation of S-nitrosothiol that retains NO bioactivity in the blood [50]. However, irreversible oxidation of this cysteine residue disintegrates the structure of Hb and, consequently, in the release of heme [51]. $\beta$Cys93 performs a role as a nitric oxide carrier and is also required for the correct oxygenation of tissues and proper cardiovascular function [52]. The increase in $\tau_c$ may be attributed to conformational changes in the Hb molecule induced by ROS or its binding to the membrane, since it was shown that mild oxidative stress initiates the binding of redox-active Hb to the membrane [53]. An increase in the correlation time of spin-labeled Hb was also observed after hemodialysis in patients with chronic kidney disease [54], where oxidative stress performs a crucial role in damaging the plasma and its components. In addition to the changes in the Hb structure, a significant decrease in the total thiols and CAT activity in the hemolysate from the varicose vein blood was observed. Free thiol groups (R-SH) are of key importance in enzyme activity in the hemolysate from the varicose vein blood was observed. Free thiol groups (R-SH) are of key importance in the plasma and its components. In addition to the changes in the plasma membrane properties, the aging process, inflammation, neurotoxicity disorders, and pesticide poisoning [63]. AChE activity in the erythrocyte membrane from varicose veins was lower in comparison to those of peripheral veins. Our research is in line with work in which oxidative stress might be responsible for a decreased AChE activity in the diaphragms of rats induced with sepsis [64]. Our obtained results suggest oxidative stress in varicose vein blood and increased ROS production in the plasma. Interestingly, oxidative stress is also found in the interior of the erythrocytes. It is possible that superoxide can be generated by damaged Hb or released heme [65]. It is also possible that ROS generated in the plasma can also penetrate the erythrocyte plasma membrane.

Thiol gets oxidized by superoxide radicals or H$_2$O$_2$ released in the plasma [66, 67]. Both molecules can penetrate through the membrane [68, 69]. Our results align with those reported by Takase et al. [44] and Bujan et al. [46], which state that neutrophils could also be activated in the plasma with the ROS generation. Other ROS can be formed via enzymatic or nonenzymatic pathways, including hypochlorous acid (HOCI) and hypobromous acid (HOBr), generated by neutrophils and eosinophils, respectively, singlet oxygen ($^1$O$_2$), nitrogen dioxide (•NO$_2$), peroxy nitrite (ONOO$^-$), and carbonate radicals (CO$_3^{1-}$) [66, 67].

Thus, the present study confirmed that, under conditions of hypoxia and inflammatory response in the blood of varicose veins, activation of phagocytic cells might occur, leading to oxidative damage to the plasma and erythrocyte components.

5. Conclusion

The results of the present study showed changes in the plasma and erythrocytes taken from varicose veins in comparison to plasma and erythrocytes taken from peripheral veins in the same patients. It was presented that conditions in the milieu of varicose vein blood lead to
oxidative damage of the plasma and erythrocytes and probably other cells in the plasma. Oxidative stress-induced structural changes in erythrocytes may affect their rheological properties, resulting in related complications in microcirculation. Moreover, for the first time, our research showed significant differences in the plasma and red blood cells properties between varicose vein blood and peripheral vein blood in the same patients; however, it is a preliminary study.

**Abbreviations**

AChE: Acetylcholinesterase  
CAT: Catalase  
CVD: Chronic venous disease  
EPR: Electron paramagnetic resonance  
GSH: Glutathione  
Hb: Hemoglobin  
HIF: Hypoxia inducible factor  
ILs: Interleukins  
ISL: Iodoacetamide spin label  
ROS: Reactive oxygen species  
TBARS: Thiobarbituric acid-reactive substance.

**Data Availability**

The datasets used and/or analysed during the current study are available from the corresponding author on reasonable request.

**Conflicts of Interest**

The authors declare that they have no conflicts of interest.

**Authors’ Contributions**

LG participated in developing the theory, planning and supervising the work, and drafting the manuscript; AP carried out the experiment, performed the statistical analysis, drafted the manuscript; JBS participated in carrying out the experiment and data analysis and drafting the manuscript; PH contributed to carrying out the experiment, designing the figures, and drafting the manuscript; KG was involved in planning and supervising the work and was the lead in writing the manuscript.

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