A calamitic mesogenic near-infrared absorbing croconaine dye/liquid crystalline elastomer composite†

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In this work, we report the first example of a calamitic mesogenic near-infrared (NIR) absorbing organic dye, made by functionalizing a thiophene–croconaine chromophore rigid core with two symmetric long flexible alkyl chains. The liquid crystal (LC) NIR dye YHD796 exhibits a sharp and intense NIR absorption band with a maximum absorption peak at 796 nm. Taking advantage of the improved solubility of YHD796 dispersed in mesogenic molecules, a homogeneously-aligned mono-domain liquid crystalline elastomer (LCE)/YHD796 composite film is successfully prepared by applying the classical LC-cell-alignment method and in situ photo-polymerization of photocurable LC monomer mixtures. This LCE/YHD796 composite film performs a fully reversible contraction/expansion response towards NIR light stimulus due to the photothermal heating effect induced by the YHD796 dye well-dispersed in the LCE matrix.

Introduction

Photoresponsive polymeric materials have garnered significant interest within the last few decades due to their ability to undergo rapid changes in response to light, which compared with other stimuli (i.e. pH, heat, electric, magnetic, chemicals, etc.), is advantageous to maximize resolution in time and space. This unique physical characteristic endows photoresponsive polymeric materials with prospective applications ranging from medical technology to robotic devices.1–5 Among them, photo-actuated liquid crystalline elastomers (LCEs) can vary their microscopic mesogenic order by absorbing photon energy to further change the macroscopic material shapes, and thus have received increasingly scientific attention.6–19

Traditional photoresponsive LCE materials relying on the trans–cis isomerization effect of incorporated azobenzene chromophores are generally light-stimulated by UV sources (λ < 400 nm). Recently, near-infrared (NIR) light sources (λ > 760 nm) have become more and more popular in photo-responsive systems because of their high biological tissue penetration ability, excellent biocompatibility and broad utility in biomedical applications.20–24 Up to now, the preparation strategies for NIR-stimulus-responsive LCEs can be divided into two categories: (1) through doping inorganic or organic up-conversion materials into azobenzene-containing LCE materials, IR irradiation can be converted into low-wavelength lights to realize azobenzene trans–cis isomerization.25,26 (2) Taking advantage of the photo-thermal effect of thermally conductive fillers (carbon nanotubes,27–30 gold nanoparticles,31–34 etc.), photons can be transformed into heat energy, which induces the LC-to-isotropic phase transition and further force the whole sample to actuate macroscopically.

Nowadays, the latter strategy dominates the field of NIR-responsive LCEs due to the benefit of versatile mesogenic structure designs unbound to azobenzene. However the most serious drawback is that most of the dopants are inorganic nanoparticles, which are usually difficult to disperse in organic LCE matrices,34 naturally weaken the matrix mechanical properties and slow down the photo-actuation response of the prepared LCE composites.34

An ideal solution could be to use organic IR absorbing dyes, which follow non-radiative relaxation pathways to realize the photo-thermal heating.

However, IR dyes are relatively rare species and most of the previously reported organic dye/LCE composites were sensitive to UV-vis light with wavelengths ranging from 300 to 650 nm.35–40 Up to now, the only two NIR absorbing dyes used in LCEs have been BASF Lumogen IR788 (ref. 41 and 42) (CAS no. 333304-54-4 (ref. 43 and 44)) and Dye 1002.43 As shown in Fig. 1, Lumogen IR788 is a quaterpylenetetracarboxylic diimide IR absorbent with a complicated polyaromatic structure, which unfortunately has a poor solubility in certain photocurable...
liquid crystal (LC) monomers based on our initial trials, while the maximum filling content of Dye 1002 was 0.2 wt% based on literature report.\textsuperscript{32} This limitation prompted us to look for an alternative IR-absorbing compound with optimal solubility in LCE matrix. Based on like-dissolve-like theory, a calamitic mesogenic NIR absorbing dye was our objective.

Our design was inspired by Smith’s croconaine dye\textsuperscript{45} (compound 1, Fig. 1) which compared with the most-studied indocyanine green (ICG, Fig. 1) dye, had “an intense absorption band, very short singlet excited state lifetime with highly efficient relaxation to the ground state, negligible intersystem crossing to triplet state and very low oxygen photosensitization ability, high chemical and thermal stability”.\textsuperscript{45} Most importantly, the thiophene-croconaine chromophore was potentially a perfect rigid mesogenic core for building calamitic LC molecular structures. Based on this design logic, we decorated the thiophene-croconaine core with two symmetric iso-nipocotates bearing different alkyl tails (–C\textsubscript{n}H\textsubscript{2n+1}, n = 1 or 10) to synthesize two LC-like NIR absorbing dyes (YHD1, YHD796, Fig. 1).

**Experimental section**

**General considerations**

The LC monomer A444 was synthesized following our previous report.\textsuperscript{44–48} All the detailed instrumentation descriptions, synthetic protocols, and NMR spectra are described in the ESL.\textsuperscript{†}

**Synthesis of 1-thiophen-2-yl-piperidine-4-carboxylic acid (8)**

Compound 6 (1.00 g 4.45 mmol) and 25 mL of 0.5 N sodium hydroxide solution were added into a 100 mL round-bottom flask. The reaction mixture was heated at reflux temperature for 3 h. After cooling to room temperature, the mixture was acidified with 10 mL of aqueous acetic acid (conc. 10%). The precipitate was collected by filtration, washed with water, and dried under vacuum.

**Synthesis of 1-thiophen-2-yl-piperidine-4-carboxylic acid decyl ester (9)**

Compound 8 (0.76 g, 3.60 mmol), n-decyl alcohol (0.68 g, 4.34 mmol), DMAP (0.02 g, 0.18 mmol) and dry CH\textsubscript{2}Cl\textsubscript{2} (25 mL) were added into a 50 mL Schlenk-type flask. Under a nitrogen atmosphere, DCC (0.88 g, 4.27 mmol) was added into the above flask in one portion at 0 °C. After filtering off the precipitate, the solution was concentrated by rotary evaporation and the crude oil was purified by column chromatography (petroleum ether : ethyl acetate = 15/1) to give the product 9.
Synthesis of 2,5-bis[(decyl-4-carboxylate-piperidylamino)thiophenyl]-croconium (YHD796)

Croconic acid 7 (0.14 g, 0.98 mmol) and compound 9 (0.69 g, 1.96 mmol) were added into a solution of 30 mL toluene/n-butanol (v/v, 1/1) under nitrogen atmosphere. The reaction mixture was heated to reflux for 1 h. After cooling to room temperature, the mixture was transferred into a 100 mL round-bottom flask and diluted with 15 mL of dichloromethane. The solution was concentrated by rotary evaporation and the resulting crude black solid was purified by column chromatography (CH$_2$Cl$_2$ : methanol = 50/1) to give the desired product YHD796 (0.64 g, yield 79.9%) as a black solid. $^1$H NMR (300 MHz, CDCl$_3$) $\delta$: 6.77 (d, $J = 6.0$, 3.0 Hz, 1H), 6.60 ($t$, $J = 6.0$ Hz, 1H), 6.13 (m, 1H), 4.16–4.03 ($t$, $J = 8.3$ Hz, 2H), 3.51 (m, 2H), 2.84 (m, 2H), 2.42 (m, 1H), 2.07–1.87 (m, 4H), 1.40–1.22 (m, 16H), 0.89 ($t$, $J = 6.6$ Hz, 3H). $^{13}$C NMR (75 MHz, CDCl$_3$) $\delta$: 174.51, 126.02, 112.31, 105.65, 64.64, 51.55, 40.57, 31.84, 29.60, 28.99, 28.60, 27.66, 25.88, 22.62, 14.04.

Preparation of LCE/YHD796 composite film

As shown in Fig. 3, LC monomer A444 (30.11 mg, 4.76 $\times$ 10$^{-2}$ mmol), hexamethylene diacylate (1.20 mg, 5.29 $\times$ 10$^{-3}$ mmol), 2,2-dimethoxy-2-phenylacetophenone (0.34 mg, 1.33 $\times$ 10$^{-3}$ mmol) and YHD796 (0.42 mg, 0.53 $\times$ 10$^{-3}$ mmol) were mixed and dissolved in CH$_2$Cl$_2$ which was then evaporated to provide the LC mixture. The mixture was filled by capillarity into an antiparallel homogeneous-aligned LC cell with 20 $\mu$m gap placed on a hot stage which set the temperature at 120 $^\circ$C. The LC cell was slowly cooled down to 60 $^\circ$C at a rate of $\sim$1.0 $^\circ$C per minute and annealed at 60 $^\circ$C for ca. 2 h to achieve a fine planar alignment. A subsequent UV irradiation was carried out under a stream of nitrogen gas (using a zip-lock bag). After UV irradiation (365 nm, 20 mW cm$^{-2}$) for 1.5 h, the LC cell was placed into a 40% hydrofluoric acid solution for 3 days to dissolve the glass cell$^{29}$ and give the desired LCE/YHD796 composite as a free-standing film.

Results and discussion

Syntheses and physical properties of NIR absorbing dyes

The synthetic protocols for the two NIR absorbing dyes (YHD1 and YHD796) are illustrated in Scheme 1. Thiophene-2-thiol (4) was first nucleophilically substituted by methyl isonipecotate (5) to give intermediate 6,$^{33}$ which was further reacted with croconic acid (7) to provide YHD1. To lengthen the alkyl tails of the croconaine dye, intermediate 6 was subjected to a sequential alkaline saponification/acid hydrolysis process and then DCC-coupled with n-decyl alcohol to synthesize its long-tail analogue 9, which was subsequently converted into the corresponding NIR dye YHD796.

The mesomorphic properties of YHD1 and YHD796 were investigated by polarized optical microscopy (POM). Unsurprisingly, the short-tail dye YHD1 with a high melting point at 236–237 $^\circ$C had no LC phase and a poor solubility in mesogenic monomer A444 (Fig. 3A).$^{56,57}$ Similar to Lumogen IR788, the mixture of A444 and YHD1 (1.0 wt%), when filled in LC cells, underwent obvious phase segregation. On the contrary, the long-tail analogue YHD796 exhibited a monotropic LC mesophase
(phase sequence and transition temperatures: crystal (K) – 84 °C – nematic [N] – 178 °C – isotropic (I) (on heating); I – 32 °C – K (on cooling)), with a characteristic nematic marble texture as shown in Fig. 2A. The nematic phase was also confirmed by wide-angle X-ray scattering analysis (WAXS, Fig. S16†) which presented no sharp scattering peaks in the low-angle region when YHD796 sample was heated to a temperature above its melting point (84 °C). Most importantly, due to the two symmetric long flexible alkyl tails, calamitic dye YHD796 could be well-solubilized in A444 system (up to ca. 10 wt%).

Examined by UV-vis spectroscopy, the mesogenic dye YHD796 exhibited a very sharp and intense NIR absorption band \( \lambda_{\text{abs,max}} = 796 \text{ nm} \), conc. = \( 2.98 \times 10^{-3} \text{ mg mL}^{-1} \) in CH\(_2\)Cl\(_2\)), which was consistent with previously reported croconaine dye analogues. Hereinafter, we focused YHD796 for further investigation and application.

**Preparation and photo-responsive behavior of LCE/YHD796 composite film**

With this novel mesogenic NIR-absorbing dye in hand, we started to dope YHD796 into photocurable LC monomer matrix and prepare the corresponding LCE/YHD796 composite films. In order to achieve a good uniaxial-alignment for synthesizing mono-domain LCE membranes, we chose the widely used LC-cell-alignment method\(^{46,54,55}\) that relied on the surface anchoring effect of anti-parallel-rubbed polyimide surface layers of LC cells to align the mesogens uniaxially. We further applied \( \text{in situ} \) photo-polymerization reaction to fabricate mono-domain LCE membranes. As shown in Fig. 3A, YHD796 was mixed with acrylate LC monomer A444, the crosslinker hexamethylenediacylate (HMDA) and the photoinitiator 2,2-dimethoxy-2-phenylacetophenone (DMPA) with a molar ratio of 1.0 : 90 : 10 : 2.5. The LC mixture was then filled into an anti-parallel surface-rubbed LC cell at the temperature above its clearing point. After lowering the temperature to the LC phase, UV-initiated (\( i = 365 \text{ nm} \) \( \text{in situ} \) photo-polymerization and photo-crosslinking were executed on the homogeneously-aligned LC mixture. Finally, the LC cell was completely etched in hydrofluoric acid aqueous solution to obtain the free LCE/YHD796 composite film. The preparation protocol is schematically illustrated in Fig. 3B, and described in details in the Experimental section.

POM technique was applied to investigate the homogeneous-alignment of the LCE/YHD796 composite film. As indicated in Fig. 4A and B, the photocurable A444/HMDA/DMPA/YHD796 mixture in a LC cell showed the lowest transmittance when the polarizer was parallel or perpendicular to the cell surface rubbing direction. When rotating the cell with an interval of 45°, maximum transmittance was clearly observed. After UV-initiated photo-polymerization, the corresponding LCE/YHD796 composite film either inside the LC cell (Fig. 4C and D) or released from HF-etched LC cell (Fig. 4E and F) also presented similar light transmittance variation behavior, although the LC birefringences were slightly different from each other in these three stages. Overall, the planar alignment of mesogenic units in mono-domain LCE/YHD796 composite film was fairly good.

For comparison, we prepared a pure LCE film containing A444/HMDA/DMPA with the exact same molar ratio (90/10/2.5) but no YHD796 dye. As demonstrated in Fig. 5, the DSC curves of the LCE/YHD796 composite and the pure LCE film both showed an enantiotropic nematic phase in a perfectly matched temperature range of \( \text{ca. } 40-106.3 \text{ °C} \), although the N-to-I

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**Fig. 4** POM images of A444/HMDA/DMPA/YHD796 mixture filled in a LC cell (A, B) before UV illumination and (C, D) after UV illumination, and (E, F) the corresponding LCE/YHD796 composite film released from HF-etched LC cell.

**Fig. 5** DSC curves of pure LCE film and LCE/YHD796 composite film during the first cooling scan and the second heating scan at a rate of 10 °C min\(^{-1}\) under nitrogen atmosphere.
enthalpy change (1.154 J g⁻¹ K⁻¹) of the LCE/YHD796 composite was slightly higher than the ΔH value (1.081 J g⁻¹ K⁻¹) of the pure LCE sample.

The optical absorption of the LCE/YHD796 composite material was examined by UV-vis spectroscopy. As shown in Fig. 2B, the composite sample had a similar NIR absorption band as the free YHD796 dye, with a maximum absorption at 796 nm when swelled in CH₂Cl₂ with a concentration of 0.11 mg mL⁻¹. Based on this, the concentration of incorporated YHD796 dye could be roughly estimated as ca. 1.44 × 10⁻⁸ mg mL⁻¹. However this value might not be accurate since YHD796 molecule is an ionic salt which might be soluble to some extent in the HF aqeous solution used during the cell-etching process, although the hydrophobic surface properties of LCE films should limit the leaching of NIRDye.

Encouraged by the UV-vis spectrum result, we used a NIR light source (output power: 8 W, center wavelength: 808 ± 3 nm) to investigate the NIR-stimulated photo-actuation characteristics of the LCE/YHD796 composite film and the pure LCE film. Theoretically, if the embedded NIR dyes could efficiently convert the absorbed photon energy into enough thermal energy to raise the local temperature of mono-domain LCE samples above their Tₙ, the mesogenic orders would change. Due to the elastic strains and forces provided by the crosslink points in the matrix, a spontaneous and reversible uniaxial shrinkage/expansion of LCE materials along the LC director should be visualized.

As indicated in Fig. 6A and B and Video S1,† the prepared LCE/YHD796 composite film underwent a rapid shrinking transformation upon NIR illumination and reached maximum contraction in 33 seconds. After NIR source was removed, the LCE composite film fully recovered its original shape. This reversible shrinkage/expansion phenomenon was observed repeatedly during several NIR-light on/off cycles. In contrary, the pure LCE film had no NIR-responsive behavior due to the lack of YHD796 dye.

In order to further confirm that the shrinking behavior of LCE/YHD796 composite film under NIR illumination was indeed induced by the photo-thermal heating effect of YHD796 dye, the surface temperature variations of LCE/YHD796 composite film and of the pure LCE film as a function of NIR irradiation time were continuously recorded by a thermal imager (FLUKE TI90). As shown in Fig. 6C, the temperature of the LCE/YHD796 sample jumped within 30 seconds from 16 °C to 115 °C, higher than its clearing point (Tₙ = 106.3 °C), while the temperature of the pure LCE film reached a maximum value of ca. 47.7 °C in the same period. These data demonstrated that the incorporated YHD796 dye was the key driving force to achieve the local anisotropic-to-isotropic phase transition in the photo-triggered actuation behavior of the LCE/YHD796 composite film.

The uniaxial expansion (L/Lₘis) of the LCE/YHD796 composite film versus NIR illumination time is plotted in Fig. 6D, where L is the length of the LCE composite film along the alignment direction at any specific irradiation time, and Lₘis is the minimum length of the material in its isotropic state. As indicated in Fig. 6D, the composite film under NIR illumination gradually realized the maximum contraction of ca. 25% in 33 seconds, and once the NIR light source was removed, it recovered its original shape in 3–5 seconds. This sharp contrast can be explained by the photo-thermal heating effect, which can only increase the local temperature of the LCE composite sample but not the environmental temperature. Thus as soon as NIR is turned off, the tiny thermal energy generated inside the LCE composite sample is spontaneously and rapidly released into the much colder and larger environment.

Conclusions

In this work, we design and synthesize a calamitic mesogenic NIR absorbing organic dye YHD796, by functionalizing a thiophene-croconaine chromophore core with two symmetric long flexible alkyl chains. To the best of our knowledge, this rod-like molecule is the first example of organic dye exhibiting not only a sharp and intense NIR absorption band with a maximum absorption peak at 796 nm but also a LC mesophase. Due to the improved solubility of YHD796 in mesogenic monomers and the remarkable photo-thermal heating effect, the corresponding LCE/YHD796 composite film prepared by using the classical LC-cell-alignment method and in situ photo-polymerization protocol, performed a fully reversible contraction/expansion response towards NIR light stimulus. Although the mesogenic NIR absorbing organic dye is obviously advantageous to synthesize well dispersed dye/LCE composites, its ionic salt form might bring some challenges when dealing with HF etching or electric field alignment process. Further development of non-ionic mesogenic NIR absorbing organic dyes are under investigation.

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