Evaluating the Thermoelectric Properties of BaTiS$_3$ by Density Functional Theory

Tula R. Paudel* and Evgeny Y. Tsymbal

ABSTRACT: BaTiS$_3$ is a semiconductor with a small bandgap of $\sim$0.5 eV and strong transport anisotropy caused primarily by structural anisotropy; it contains well-separated octahedral columns along the [0001] direction and low lattice thermal conductivity, appealing for thermoelectric applications. Here, we evaluate the prospect of BaTiS$_3$ as a thermoelectric material by using the linearized electron and phonon Boltzmann transport theory based on the first-principles density functional band structure calculations. We find sizable values of the key thermoelectric parameters, such as the maximum power factor PF = 928 $\mu$W K$^{-2}$ and the maximum figure of merit ZT = 0.48 for an electron-doped sample and PF = 74 $\mu$W K$^{-2}$ and ZT = 0.17 for a hole-doped sample at room temperature, and a small doping level of $\pm$0.25e per unit cell. The increase in temperature yields an increase in both the power factor and the figure of merit, reaching large values of PF = 3078 $\mu$W K$^{-2}$ and ZT = 0.77 for the electron-doped sample and PF = 650 $\mu$W K$^{-2}$ and ZT = 0.62 for the hole-doped sample at 800 K. Our results elucidate the promise of BaTiS$_3$ as a material for the thermoelectric power generator.

INTRODUCTION

The thermoelectric heat–current conversion process is inefficient compared to other energy conversion technologies such as fossil fuel, solar, and nuclear because of the small thermoelectric figure of merit, ZT, of the materials involved. While a large ZT value of $\sim$3 to 4 is required for thermoelectric technology to be viable, even optimized thermoelectric bulk materials, including doped SnTe and PbTe,$^{1-4}$ have ZT values not exceeding about 2.6, which limits the usage of thermoelectricity in practice as a primary energy source. Alternatively, thermoelectric power generation can supplement the other energy technologies by converting wasted heat into current, thereby increasing the overall efficiency of the system. These thermoelectric power-generating devices are very reliable energy producers as they are free of lifetime-reducing moving parts (the typical lifetime of a thermoelectric device is $\sim$10$^6$ h) and serve as a component of a global energy solution system.

The thermoelectric figure of merit, ZT = $(S^2\sigma/\kappa)T$, is a key measure of a material’s thermoelectric performance at temperature $T$, where $S$, $\sigma$, and $\kappa$ represent the thermoelectric power, electrical conductivity, and thermal conductivity, respectively. The latter is given by the sum of electronic thermal conductivity, $\kappa_e$, and lattice thermal conductivity, $\kappa_l$. The thermoelectric performance of a material can be enhanced by increasing ZT, which can be achieved by maximizing the power factor, PF = $S^2\sigma$, and minimizing $\kappa$.$^5$ Unfortunately, maximizing PF, for instance, by band engineering,$^6$ often also increases $\kappa_e$, even though the contrary has been reported$^7$ a significant part of $\kappa$, as it is connected to $\sigma$ by the Wiedemann–Franz law, $\kappa_e\sigma = LT$, where $L$ is a constant equal to 2.44 $\times$ 10$^{-8}$ V$^2$ K$^{-2}$ for a free-electron gas. Fortunately, $\kappa_l$ can be minimized somewhat independently without interfering with the process of power factor $S^2\sigma$ maximization, thereby optimizing the overall ZT. $\kappa_l$ can be reduced by using a low-dimensional form of materials, including interfaces, nanostructures, grain boundaries, and amorphous powder, in which the lattice coherency can be tuned, and in the ideal case made smaller than the heat-carrying-phonon mean free path, thereby increasing the phonon scattering.

RESULTS AND DISCUSSION

BaTiS$_3$ is a unique compound of the II-IV-S$_3$ material class that crystallizes in the hexagonal P6$_3$mc phase (space group 186, Figure 1a), while other members of this class, such as Ba(Zr,Hf)S$_3$ and Ba(Zr,Hf)Se$_3$, crystallize in the orthorhombic Pnma phase. BaTiS$_3$ contains one-dimensional edge-sharing octahedral pillars, running along the [0001] and [0100] directions,$^8$ that are connected by the van der Waals interaction along the [1000] and [0010] directions. This very anisotropic structure gives rise to giant optical anisotropy, broadband birefringence
and dichroism,\(^1\) and ultralow thermal conductivity.\(^10\) The electronic band structure and electronic density of states (eDOS) show that the top of the valence band contains two-fold degenerate in-plane sulfur \(p_x\) and \(p_y\) bands (Figure 1b). The bottom of the conduction band contains nondegenerate titanium \(d_3^z\) bands. The spin–orbit coupling does not affect the conduction band, as \(I_s = 0\) for the \(d_{3z^2}\) band, and is very small for the valence band, as it is primarily composed of low-atomic-number \(s\)-\(p\)-like states. The Fermi surface of the doped BaTiS\(_3\) \((\pm 0.05\ e/\text{uc})\) represents a \(\Gamma\)-point-centered Fermi ellipse with the major axis pointing along the out-of-plane direction for a doped-electron sample (Figure 1d) and along the in-plane direction for a hole-doped sample (Figure 1c).

Upon increasing the doping level to 0.5 \(e/\text{uc}\), while the Fermi surface of the hole-doped sample extends in the plane of the Brillouin zone with multivalley degeneracy at the \(K\) and \(K'\) points (Figure 1e), which leads to large DOS and Brillouin zone with multivalley degeneracy at the \(\Gamma\)-\(Z\) direction (Figure 1f), suggesting a low-dimensional nature of the band structure and possibly large \(\mu\).

The effective mass of the undoped BaTiS\(_3\) (calculated using the band curvature \(1/m^* = 1/K^2(\partial^2E/\partial k^2)\)) shows that the smallest electron mass is 0.7\(m_0\) along the out-of-plane direction and the smallest hole mass is 0.8\(m_0\) along the in-plane direction, where \(m_0\) is a free-electron mass (Table 1). The small effective mass indicates the possibility of high mobility. Indeed, the mobility calculated using a single-band deformation model\(^11\) shows a very large value of 1240 \(V\ \text{cm}\ \text{s}^{-1}\) \(\text{eV}^{-1}\), similar to TiS\(_2\)\(^12\) for electron conduction and 90 \(V/(\text{cm S})\) for hole conduction. Large mobility suggests high electrical conductivity, \(\sigma = ne\mu\), and a long carrier lifetime, \(\mu = e\tau/m^*\), for electron conduction.

Based on the large eDOS in the hole-doped sample and large \(\mu\) in the electron-doped sample, PF is expected to be significant for both electron and hole doping. A combination of large PF and low \(\kappa_s\) suggests that BaTiS\(_3\) could be an excellent material for thermoelectric applications. To test this expectation, we explicitly compute \(\sigma/\tau\), \(S\), and \(\kappa_s/\tau\) using the linearized electron Boltzmann transport theory based on the first-principles density functional band structure calculations assuming constant relaxation time, and \(\kappa_s\) using the corresponding phonon counterparts assuming relaxation time based on three-phonon interactions.\(^13\)

**Figure 2a**–c shows total eDOS, doping level, \(n\), and out-of-plane and in-plane components of \(\sigma\), \(S\), and \(\kappa_s/\tau\) as a function of the chemical potentials, \(\mu = E - E_F\), of the chemical potential, \(\mu = E - E_F\). The chemical potentials we choose lie within the reasonable doping level of \(\pm 0.65\ e/\text{uc}\). As expected for a semiconductor, the eDOS, the doping \(n\), and the conductivities, \(\sigma\) and \(\kappa_s\), are zero at the Fermi energy; they, however, increase quickly with the increasing \(\mu\), which can be brought about by doping with electrons or holes, as shown in Figure 2a (red curve). The temperature additionally increases.

| type | \(E_i\) (eV) | \(C_{ip}\) | \(C_{op}\) (Pa) | \(m^* (m_0)\) | \(\mu (\text{cm}^2\text{V}^{-1}\text{s}^{-1})\) | \(\tau (\text{ps})\) |
|------|------------|----------|---------------|----------------|------------------|---------|
| \(e\) (ip) | 7.8 | 942 | 1.6 | 29 | 0.03 |
| \(h\) (ip) | -11.2 | 942 | 0.8 | 80 | 0.04 |
| \(e\) (op) | -3.9 | 1266 | 0.7 | 1240 | 0.49 |
| \(h\) (op) | 5.6 | 1266 | 1.5 | 90 | 0.07 |

**Figure 1.** Atomic and electronic structures of bulk BaTiS\(_3\). Atomic structure (a), electronic band structure (left panel) and atom- and orbital-resolved density of states (right panels) (b). High symmetry points \(\Gamma\), \(Z\), \(K\), and \(M\) are marked in the Brillouin zone shown in (c). Fermi surface of hole-doped BaTiS\(_3\) with concentrations of 0.05 \(e/\text{uc}\) (c) and 0.5 \(e/\text{uc}\) (e) and that of electron-doped BaTiS\(_3\) with concentrations of 0.05 \(e/\text{uc}\) (d) and 0.5 \(e/\text{uc}\) (f).
the conductivity, both electronic and thermal (as can be seen by comparing Figure 2 to Figure S2), typical for a semiconductor. While the conductivity of the hole-doped sample remains mostly isotropic, the conductivity of the electron-doped sample is very anisotropic: the out-of-plane conductivity increases dramatically compared to the in-plane conductivity due to contribution from very dispersive out-of-plane Ti-d$_z^2$ bands. Upon increasing the electron doping level beyond 0.65 e/uc, the out-of-plane d$_{x^2−y^2}$ and d$_{xz}$ bands contribute first to the conductivity (Figure 1b), and eventually, the in-plane bands, d$_{3z^2−r^2}$ and d$_{xy}$, start to participate and the conductivity increases up to 2100 m$Ω^{-1}$ cm$^{-1}$ (1 $Ω^{-1}$ m$^{-1}$ = 1666 $μΩ$ cm$^{-1}$, not shown in the plot). The in-plane component, however, remains small ($\sim$1−15 m$Ω^{-1}$ cm$^{-1}$), giving rise to a very large conduction anisotropy, $\sigma_{\text{max,ip}}/\sigma_{\text{max,op}}$, of 225, consistent with the experimental observations.

The Seebeck coefficient, $S$ (also called thermopower), which is proportional to the logarithmic derivative of conductivity with energy, shows a usual sign change at $\mu_c = 0$ with a positive value for hole doping and a negative value for electron doping. For the electron doping, the thermopower has a maximum value of $\sim$735 $μV$ K$^{-1}$ for the out-of-plane component and $\sim$666 $μV$ K$^{-1}$ for the in-plane component. Similarly, for the hole doping, thermopower has a maximum value of $\sim$689 $μV$ K$^{-1}$ for the out-of-plane component and 733 $μV$ K$^{-1}$ for the in-plane component at room temperature. These values are much larger than 150−250 $μV$ K$^{-1}$ associated with good bulk thermoelectric materials. $^{14}$ When the temperature is increased, $S$ decreases; however, it remains relatively high, with maximum values of 390 $μV$ K$^{-1}$ for a hole-doped system and 381 $μV$ K$^{-1}$ for an electron-doped system at 800 K (Figure S2), around which temperature the compound is reported losing $S$ and doping the system n-type. $^{13}$ At higher temperature, typically, the most significant peak of $S$ shifts to a higher doping level. However, here it is achieved at a low doping level, $\sim$0.01 e/uc, which is much lower than that in its oxide counterpart, electron-doped SrTiO$_3$. $^{16}$

Figure 3a shows the calculated in-plane and out-of-plane $κ_l$ plotted as a function of temperature. It can be seen that $κ_l$ is also anisotropic (Figure 3a), similar to $σ$ (Figure 2b), with the out-of-plane component of 2.35 V m$^{-1}$ K$^{-1}$ and the in-plane component of 0.25 V m$^{-1}$ K$^{-1}$ at 300 K. The value of $κ_l$ is larger, comparable, and smaller than the electronic thermal conductivity, $κ_e$, for intrinsic, low-doped, and heavily doped samples, respectively (compare Figure 3a to Figure 2d). Thus, the control of $κ_l$ is crucial for the ZT of intrinsic and low-doped semiconductors. Even the largest values of the calculated $κ_l$ for BaTiS$_3$ are smaller than those reported for the other two-dimensional (2D) materials such as phosphorene, MoS$_2$, and graphene but comparable to stannate and n-doped STO.$^{17}$

The efficiency of a thermoelectric generator is related to the power factor, $PF = S^2\sigma$. Figure 4a shows $PF$ plotted as a function of the chemical potential, $μ_c = E − E_F$. $\mu_c$ is also strongly anisotropic, similar to $S$ and $σ$. At room temperature, the maximum power factor is 928 $μW$ K$^{-2}$ cm for an electron-doped sample and 74 $μW$ K$^{-2}$ cm for the hole-doped sample. With increasing temperature, the power factor increases and reaches a value of 3078 $μW$ K$^{-2}$ cm for the electron-doped sample and 650 $μW$ K$^{-2}$ cm for the hole-doped sample at 800 K (Figure S3), which is largely contributed by the enlarged $σ$ at the higher temperature (compare Figure 2b to Figure S2b).

The reason for the small value of $κ_l$ can be gleaned from the computed phonon spectra. Figure 3b shows the computed phonon band structure and phonon density of states (pDOS), and Table 2 shows the frequency of the phonons at the center of the Brillouin zone, which are categorized according to their symmetries within the underlying point group $C_{6v}$. We see from the phonon spectra that the optical phonons are soft, and there is no gap between the acoustic and optical phonons. There are several reasons for this, including a weak ionic...
The doping level that is required for the maximum value of ZT (0.77) is small and can be easily achieved by doping. For example, S vacancies or doping by La (similar to the case of well-known LaBa1−xMnO3) can make the compound n-type with a good thermoelectric figure of merit. P-doping can be induced by high-temperature growth at a reduced pressure of Ba. The predicted large thermoelectric efficiency of BaTiS3 for both electron- and hole-type conduction is due to the relatively low bandgap and possibly low ionization potentials compared to oxides. Further tuning of the band structure and thermal conductivity is possible by phase intermixing and exfoliations. We find that the alternative centroсимmetric P63/mmc (space group 194) phase is unfavorable just by ΔE ≈ 1 meV Å−1, indicating a possibility for intermixed phases. The calculated cleavage energy of 1.1 J m−2 for exfoliation of the compound along the [0001] direction, which is just about three times the corresponding values for MoS2 and graphene, indicates the possibility of creating low-dimensional ternary chalcogenides.

### COMPUTATIONAL METHODS

We used the first-principles density functional theory calculations based on the projected augmented wave method16 and the Perdew–Burke–Ernzerhof functional,19 as implemented in the Vienna ab initio simulation package (VASP).20,21 We fully relaxed the structure, the P63/mmc (space group 186) phase of BaTiS3 (Figure 1a), with the force convergence limit of 0.005 eV/atom. We treated the correlation effects beyond generalized gradient approximation (GGA) at a semimtrial GGA + U level within a rotationally invariant formalism with U = 6 eV for the Ti 3d orbitals. The crystal structure was optimized, resulting in lattice parameters a = b = 6.73 Å and c = 5.92 Å, which are very close to the experimental lattice parameters a = b = 6.75 Å and c = 5.8 Å.8 For the accurate description of the interlayer separation, van der Waal’s interaction was included in the calculations.23

Phonon band structure was calculated using the density functional perturbation theory, as implemented in the PHonon package of Quantum ESPRESSO. We used 6 × 6 × 6 Monkhorst k-grid to calculate the dynamical matrix. The interatomic force constant in real space was then calculated by Fourier transforming the dynamical matrix calculated in a uniform grid. The phonons at any other q-points were then calculated by redoing the Fourier transformation. For the electronic density of states, we used a 12 × 12 × 10 Monkhorst k-point mesh.

The conductivities, σ and κν, and the Seebeck coefficient, S, were calculated using the BOLTZTRAP code,24 which assumes a constant carrier relaxation time, τc, and can directly calculate the power factor PF = S2σ.

We estimated the carrier relaxation time, τc, by single-band approximation using

$$\tau_c(T) = \frac{\mu m^*}{e} = \frac{m^* e \hbar^2 C_j}{3k_B T |m^*|^2 E_i^2}$$

where Cj’s are the elastic modulus, m* is the effective mass, m* = h^2(∂^2E/∂k^2)^−1, h is Planck’s constant, k_i is the magnitude of the wave vector in the i-direction, T is the temperature, and E_i is the deformation potential. E_i is proportional to the band edge (conduction band minimum (CBM) and valence band maximum (VBM)) shifts, ΔE, induced by lattice dilation, Δl/l, along the crystallographic direction, i, as ΔE = E_i(Δl/l). While C_j’s were calculated using

| Table 2. Calculated Born Effective Charge Tensor Components; S1 Ion is Closer to Ba than S2 |
|-----------------|-----------------|-----------------|
|                  | xx              | yy              | zz              |
| Ba               | 3.4             | 3.4             | 3.0             |
| Ti               | 2.4             | 2.4             | 6.7             |
| S1               | −1.2            | −2.7            | −3.2            |
| S2               | −2.3            | −1.6            | −3.2            |

Large ionic screening, as follows from the large values of the dielectric permittivity tensor components (Table 3), and weak interatomic force constants. The nondispersive phonons at the Brillouin zone edges (M and K points) dominate PDOS, which allow for the larger number of phonon interactions responsible for the reduced lattice thermal conductivity. The anisotropy of the lattice thermal conductivity can be qualitatively explained in terms of the anisotropic bonding in the compound. While there is a bonding between Ti and S atoms along the out-of-plane directions, leading to relatively dispersive phonon bands (not shown) with smaller mass, larger velocity, and larger conductivity, the absence of bonding along the in-plane direction leads to the opposite: flatter phonon bands (Figure 3a) with larger mass, smaller velocity, and smaller conductivity.

The better performance measure of thermoelectric materials is a figure of merit, ZT. Figure 4b shows ZT as a function of the chemical potential μc at 300 K, which is calculated assuming k_i is independent of μc. For the electron doping, the ZT has a maximum value of 0.48 for the out-of-plane component and 0.05 for the in-plane component. Similarly, for the hole doping, ZT has a maximum value of 0.17 for the out-of-plane component and 0.16 for the in-plane component. ZT increases with the temperature because of the increase in the power factor close to μc = 0. For example, for the electron doping, ZT reaches the maximum value of 0.77 for the out-of-plane component and 0.4 for the in-plane component, and for the hole doping, it reaches a maximum value of 0.61 for the out-of-plane component and 0.40 for the in-plane component at 800 K (Figure S3).

| Table 3. Calculated Dielectric Permittivity Tensor Components |
|-----------------|-----------------|-----------------|-----------------|
|                  | xx              | yy              | zz              |
| electronic      | 10              | 10              | 13              |
| ionic           | 19              | 19              | 92              |
| total           | 29              | 29              | 105             |

**Article**

ACS Omega 2020, 5, 12385−12390

https://dx.doi.org/10.1021/acsomega.0c01139

ACS Omega 2020, 5, 12385−12390
the density functional perturbation theory.\textsuperscript{25} $E_i$’s were determined by performing a series of calculations by strain- ing a lattice along the in-plane and out-of-plane crystallographic directions.

The lattice thermal conductivity, $\kappa$, was calculated from the solution of the linearized phonon Boltzmann equation,\textsuperscript{26} assuming its proportionality with the phonon lifetime, which is determined by first-principles anharmonic lattice dynamics\textsuperscript{13} using the Phono3py code. The force constants used in the code were computed using VASP with a $2 \times 2 \times 2$ supercell and $6 \times 6 \times 6$ k-points.

**Phonon Symmetry.** The phonons at the $\Gamma$ point can be written as $\Gamma = 4A_\Gamma \oplus A_\Sigma \oplus 4B_\Sigma \oplus B_\Gamma \oplus 10E_\Gamma \oplus 10E_\Sigma$, where numeric symbols correspond to the phonon count. Out of 30 modes corresponding to 10 atoms in a unit cell, three modes are acoustic type and contain a degenerate $E_1$ (Figure S1c) mode corresponding to the vibration of the lattice along two in-plane lattice vectors’ direction and a nondegenerate $A_1$ (Figure S1a) corresponding to the vibration of the lattice along the out-of-plane lattice vector direction. The remaining modes are optical. The lowest transverse optical mode has an $A_3$ (Figure S1b) symmetry and includes an in-plane rotation of $S$ atoms around Ba. This mode is even with respect to all symmetry operations of the group, and the $E_2$ mode (Figure S1d), antisymmetric with respect to $C_{\sigma}$ and $C_3$ rotations, but odd with respect to $\sigma_c$ (contains the main rotation axis) and $\sigma_d$ reflections (does not contain rotation, still vertical) and silent with respect to both infrared and Raman spectroscopy. Out of the rest, the $A_1$ mode, symmetric with respect to all symmetry operations of the group, and the $E_2$ mode, symmetric with respect to $C_{\sigma}$ but antisymmetric with respect to $C_3$ and $C_{\nu}$ are Raman and infrared active. The $E_2$ mode (Figure S1d), antisymmetric with respect to $C_{\sigma}$ and $C_3$ but symmetric with respect to $C_{\nu}$ is only Raman active. The $B_1$ mode (Figure S1e), symmetric with respect to $C_3$ and $\sigma_c$ but antisymmetric with respect to $C_{\sigma}$ $C_3$, and $\sigma_d$ and the $B_2$ mode (Figure S1f), symmetric with respect to $C_3$ and $\sigma_d$ but antisymmetric with respect to $C_{\sigma}$ $C_3$, and $\sigma_d$ are both Raman as well as infrared inactive similar to $A_2$.

### ASSOCIATED CONTENT

**Supporting Information**

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsomega.0c01139.

Figure S1: Symmetry-resolved phonon modes. Figures S2 and S3: calculated total electronic density of states, doping concentration, in-plane and out-of-plane components of electronic conductivities, Seebeck coefficients, electronic part of thermal conductivities, power factor, and thermoelectric figure of merit at 800 K plotted as a function of the chemical potentials. Figures S4 and S5: the temperature dependence of the density of states, in-plane and out-of-plane components of electrical conductivity, Seebeck Coefficient, electronic part of thermal conductivity, power factor, and thermoelectric figure of merit plotted as a function of the chemical potentials. Table S1: computational cell parameters and atomic positions, a part Quantum Espresso input file (PDF)

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**Notes**

The authors declare no competing financial interest.

### ACKNOWLEDGMENTS

This work was supported by the Nebraska Public Power District through the Nebraska Center for Energy Sciences Research at the University of Nebraska-Lincoln, NCESR grant number 19-SE-2018. Computations were performed at the University of Nebraska Holland Computing Center. The atomic structure was produced using VESTA software.\textsuperscript{27}

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