OBJECTIVE: The present investigation is based on the green synthesis of copper nanoparticles (CuNPs) from aqueous extract of *Capparis spinosa* L. fruit. Their effects on liver function and hematological parameters in mice were evaluated.

**MATERIALS AND METHODS:** The green synthesis of CuNPs by means of *C. spinosa* extract was achieved. Ultraviolet-visible spectroscopy, fourier transform infrared (FTIR) spectroscopy, scanning electron microscopy (SEM), and energy dispersive X-ray spectroscopy were used to identify the synthesized nanoparticles. BALB/c mice were orally administrated CuNPs at doses of 1000, 2000, and 5000 µg/kg for 2 weeks. Later, the effects of CuNPs on liver function in the treated mice were evaluated by measuring the serum levels of enzymes such as alanine aminotransferase, aspartate aminotransferase, alkaline phosphatase, and bilirubin as well as hematological parameters including hemoglobin, hematocrit, white blood cell, red blood cell, and platelet counts.

**RESULTS:** A maximum peak at wavelength 414 nm confirmed the biosynthesis of CuNPs. FTIR spectrum analysis revealed that the factor groups shaped a coating extract on the surface of the nanoparticles. SEM images demonstrated a particle size between 17 and 41 nm. Although some liver enzymes and hematological parameters increased with increasing dose of extract, there was no significant difference (p>0.05) between oral administrations of CuNPs at doses of 1000, 2000, and 5000 µg/kg and the control group.

**CONCLUSION:** The findings revealed that CuNPs biosynthesized from aqueous extract of *C. spinosa* fruit have no toxic effects on the liver functions and hematological parameters of mice. However, more studies are needed for evaluation of the hepatoprotective effects of CuNPs.

**KEY WORDS:** Nanoparticles, copper, liver, hematology, *Capparis spinosa*, BALB/c mice

**ABSTRACT**

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INTRODUCTION

Nanotechnology is one of the most useful technologies and it can be applied in many areas including food and nutrition, biomedical science, gene transmission, energy science, electronics, and the space industry. In particular, this technology is implemented in the treatment of cancer, allergies, inflammation, diabetes, and other diseases.¹

There are various physical and chemical methods for the production of nanoparticles, and they are still being investigated for the purpose of obtaining particles with a certain size and lower toxicity.² Green synthesis is considered a new approach to prevent the production of undesired or unsafe by products via reliable, maintainable, and eco-favorable synthesis techniques. Between the current green procedures of synthesis of nanoparticles, use of plant extracts is an appropriate and easy method to harvest nanoparticles on a large scale relative to bacteria and/or fungi mediated synthesis.³⁻⁴ Recent studies have also shown that the synthesis of metal nanoparticles using plant extracts, i.e. green synthesis, has some benefits such as low cost and low toxic effects for large-scale production.²⁻⁴

Copper (Cu) is one of the most useful elements in medical science because of its numerous anti-inflammatory, anticancer, analgesic, and antimicrobial effects.⁵ In recent years, it has been proven that, because of their high surface-to-volume ratio, Cu nanoparticles (CuNPs) are extremely reactive and simply interact with other particles, leading to a wide range of biological activities.⁵⁻⁷

Previous reviews on laboratory animals have demonstrated that the liver is considered the key target tissue of drug toxicity. Hence, assessing the function of this organ is a very important method to determine drug toxicity.⁸ Nowadays, one of the main criteria to determine liver damage is measuring serum levels of enzymes such as alanine aminotransferase (ALT), aspartate aminotransferase (AST), alkaline phosphatase (ALP), and bilirubin.⁹⁻¹¹ It has been reported that oral consumption of some medicinal agents may affect the hematological parameters and result in anemia, neutropenia, and thrombocytopenia; therefore, it is necessary to evaluate the effects of novel medicinal drugs by measuring hematological parameters.¹² According to the present study examined by KHATAMI et al. Toxicity Effects of Copper Nanoparticles according to a method described elsewhere. Briefly, 75 mL of the obtained extract was added to 100 mL of 0.01 M Cu sulfate solution. After stirring, it was kept at 60°C for 1 day. In the next step, to remove all impurities, it was centrifuged twice at 12,000 rpm for 20 min. Nanoparticles started to deposit when the color of the solution changed from green to amber yellow. The synthesized nanoparticles were heat treated in an oven at 60°C for further analyses.

Ultraviolet-visible (UV-Vis) spectroscopy analysis

Transformation of the Cu ions to CuNPs was shown by the surface plasmon resonance (SPR) of the CuNPs. For this purpose, 0.3 mL of the specimens was diluted with 3 mL of normal saline and studied via UV-Vis spectrum analysis by means of a spectrophotometer (JENWAY 6405) in the range of 300-700 nm.¹³

Fourier transform infrared (FTIR) spectroscopy

After pouring and mixing the obtained samples and potassium bromide granules together in a ratio of 1 to 100 (1/100 ratio) and compacting them into tablets, FTIR (model Nicolet 32) analysis was carried out in the range of 400-4000 cm⁻¹ and with the resolution of 1-4 cm⁻¹.¹⁴

Scanning electron microscope (SEM)

To obtain the characteristics of the synthesized nanoparticles, electron microscopy (Mira3, Czech Republic) with 15 kV, magnification of 10x, and resolution of 1 nm was performed.

Animals and study design

A total of 32 male BALB/c mice weighing 25-30 g were provided by the Tehran Pasteur Institute and kept in light/dark cycles of 12:12 h. The room temperature was 22±2°C and the mice had ad libitum access to water and food. They were placed in laboratory conditions 30 min before the start of the experiment. The ethical approval required for this study (IR. LUMS.REC.1398.247) was issued by the Ethics Committee of Lorestan University of Medical Sciences, Lorestan, Iran. The mice tested were assigned to the following four groups:

Group i: Received normal saline orally for 14 days;
Group ii: Received CuNPs at a concentration of 1000 µg/kg orally for 14 days;
Group iii: Received CuNPs at a concentration of 2000 µg/kg orally for 14 days;
Group iv: Received CuNPs at a concentration of 5000 µg/kg orally for 14 days.

Sample collection

On day 15 of the experiments, the mice were anesthetized by ketamine-xylazine followed by collection of blood samples from each mouse after the heart was opened. The collected blood samples were put into tubes with or without anticoagulant
to process their clots and then their sera were separated by centrifugation at 5000x g for 10 min.

Evaluation of the serum liver enzymes
To determine the hepatoprotective effects of CuNPs, different clinical chemistry parameters related to liver function such as AST, ALT, ALP, and bilirubin (direct and total) were assayed by commercial diagnostics kits (Roche, Germany).10,15

Hematological parameters
To assess the effects of CuNPs on hematological studies, total collected blood was put into tubes containing ethylenediaminetetraacetic acid. Next, hematological parameters including hemoglobin, hematocrit, white blood cell, red blood cell, and platelet (PLT) counts were measured by Sysmex (KX-21, Japan).

Statistical analysis
SPSS version 17.0 (SPSS Inc., Chicago, IL, USA) was used to analyze the obtained data. One-Way ANOVA and Tukey’s post-hoc test were performed to assess the difference between the experimental groups and p<0.05 was considered statistically significant.

RESULTS

Ultraviolet-visible spectrum analysis
The results obtained showed that the highest peak of the synthesized CuNPs was found at a wavelength of 414 nm. The characteristic of the resonance band of the SP happened for CuNPs at 414 nm wavelength (Figure 1).

Fourier transform infrared analysis
The FTIR findings revealed that the biomolecules in the extract decreased the Cu sulfate solution. The bands at 3380, 2928, 1741, 1604, 1400, 1050, and 1271 were indexed to the O-H stretching of alcohol and phenol, C-H stretching of the aliphatic group, C=O stretching of ester carbonyl, C=C stretching of the aromatic ring, and C-O stretching of ester, respectively (Figure 2).

Scanning electron microscope analysis
Following the confirmation of the synthesized nanoparticles through color modification and Vis-UV and FTIR, the characterization of nanoparticles was determined with SEM. As shown in Figure 1, the synthesized CuNPs represent spherical morphology, whereas the size of the particles was between 17 and 41 nm (Figure 3).

Hepatoprotective effects of CuNPs
Table 1 shows the results of hepatoprotective effects on serum biochemical parameters in mice receiving CuNPs at doses of 1000, 2000, and 5000 µg/kg for 14 days. As can be seen, although these parameters increased with increasing doses of extract, there was no statistically significant difference (p>0.05) between oral administrations of CuNPs at doses of 1000, 2000, and 5000 µg/kg and the control group.
**Effect on hematological parameters**

As shown in Table 2, following the oral administrations of CuNPs at the employed doses of 1000, 2000, and 5000 µg/kg for 14 days, there was no significant difference (p<0.05) between hematological parameters compared with the control group.

**Table 1. Effects of CuNPs on serum liver enzymes after 2 weeks administration in mice. The results are given as mean ± standard deviation**

| Clinical biochemistry parameters | AST (U/L) | ALT (U/L) | ALP (U/L) | TB (mg/dL) |
|---------------------------------|-----------|-----------|-----------|------------|
| Control                         | 130.6±8.15| 38.3±4.31 | 132.3±8.15| 0.12±0.03  |
| CuNPs (1000 µg/kg)              | 142.4±6.15| 40.3±3.15 | 139.6±6.5 | 0.15±0.05  |
| CuNPs (2000 µg/kg)              | 138.8±5.51| 39.3±2.51 | 142.3±5.01| 0.14±0.04  |
| CuNPs (5000 µg/kg)              | 144.3±8.15| 42.7±4.36 | 144.6±6.15| 0.17±0.05  |

CuNPs: Copper nanoparticles, ALT: Alanine aminotransferase, ALP: Alkaline phosphatase, AST: Aspartate aminotransferase, TB: Total bilirubin

**DISCUSSION**

In recent years, studies in the field of nanotechnology have demonstrated that physical and chemical procedures to create nanoparticles, despite having exceptional biological activities, due to having some restrictions like toxicity, are replaced by a wide range of methods such as green synthesis.1 Today, considering that nanoparticles are broadly used by a large proportion of the world’s population to treat some diseases, it is necessary to measure the toxicity of these products by different methods. Since recent studies have shown that CuNPs possess a wide range of biological activities, we decided to evaluate their toxicity and hepatoprotective effects in mice.

Recent studies have shown that liver enzyme measurement is one of the key diagnostic tests to evaluate the liver function and also inflammations and damage such as hepatitis and cirrhosis.8,10 Therefore, they have numerous biological activities.7,18 It has been proven that some medications can cause irreparable complications through reducing white and red blood cells and the number of blood PLTs; thus, it is necessary to evaluate the effects of novel medicinal drugs by measuring hematological parameters.19

Based on the results of the present study, after oral administration of mice with CuNPs at doses of 1000, 2000, and 5000 µg/kg for 14 days, although some parameters increased with increasing dose of extract, there was no statistically significant difference (p<0.05) between oral administrations of CuNPs at these doses and the control group. Moreover, no statistically significant difference (p<0.05) was observed in hematological parameters between mice treated with CuNPs and the control group.

Considering the study of the hepatoprotective effects of nanoparticles and similar to our findings, Zhang et al.20 demonstrated that silver nanoparticles synthesized using *Rhizophora apiculata* were effective in protecting the liver from harm induced by carbon tetrachloride.

Bhuvaneswari et al.21 reported that increased enzymatic levels of AST, ALT, ALP, and bilirubin by CCl4 returned to normal when treated with silver nanoparticles synthesized using fruit extract of *Emblica officinalis*, indicating potent hepatoprotective effects of these nanoparticles. In the study conducted by Eftekhari et al.22, quercetin nanoparticles showed remarkable hepatoprotective activity via decreasing levels of AST, ALT, and ALP. Ghosh and Gomes23 also recently showed that gold nanoparticles synthesized by *Trigonella foenum-graecum* extract significantly normalized the increased enzymatic levels of AST, ALT, ALP, and bilirubin induced by CCl4, which indicated the hepatoprotective potential of these nanoparticles.

In the present study, we applied *C. spinosa* to facilitate the synthesis of CuNPs. Next, we synthesized CuNPs that were spherical and ranging from 17 and 41 nm in size. To date, a wide range of plants such as *Syzygium aromaticum*, *Nerium oleander*, *Capparis zeylanica*, *Linn.*, *Nerium oleander*, *Syzygium aromaticum*, *Embilica officinalis*, *Capparis zeylanica*, *Gloriosa superba*, and *Vitis vinifera* have been applied in the biosynthesis of CuNPs. However, identifying the plant’s capacity as biological material for the synthesis of nanoparticles in full detail requires more investigations.5,6,17,24-27

**CONCLUSION**

The findings revealed that CuNPs biosynthesized from aqueous extract of *C. spinosa* fruit have no toxic effect on the liver of the studied mice, and no significant toxicity was observed in their hematological parameters. However, more studies need to be done for evaluation of the hepatoprotective effect of CuNPs.

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