Radioisotopes Demonstrate the Contrasting Bioaccumulation Capacities of Heavy Metals in Embryonic Stages of Cephalopod Species

Thomas Lacoue-Labarthe1,2, Roger Villanueva3, Claude Rouleau4, François Oberhänsli1, Jean-Louis Teyssié1, Ross Jeffree3,5, Paco Bustamante2

1 International Atomic Energy Agency – Environment Laboratories, Monaco, 2 Littoral Environnement et Sociétés (LIENSs), UMR 6250 CNRS – Université de La Rochelle, La Rochelle, France, 3 Institut de Ciències del Mar, CSIC, Passeig Marítim de la Barceloneta 37–49, Barcelona, Spain, 4 Institut Maurice-Lamontagne, Mont-Joli, Canada, 5 Faculty of Science, University of Technology Sydney, Sydney, Australia

Abstract

Cephalopods play a key role in many marine trophic food webs and also constitute alternative fishery resources in the context of the ongoing decline in finfish stocks. Most coastal cephalopod species of commercial importance migrate into shallow waters during the breeding season to lay their eggs, and are consequently subjected to coastal contamination. Eggs of common cuttlefish Sepia officinalis, European squid Loligo vulgaris, common octopus Octopus vulgaris and the sepiolid Rossia macrosoma were exposed during embryonic development to dissolved 110mAg, 109Cd, 60Co, 54Mn and 65Zn in order to determine their metal accumulation efficiencies and distribution among different egg compartments. Cuttlefish eggs, in which hard shells enclose the embryos, showed the lowest concentration factor (CF) values despite a longer duration of exposure. In contrast, octopus eggs, which are only protected by the chorionic membrane, accumulated the most metal. Uptake appears to be linked to the selective retention properties of the egg envelopes with respect to each element. The study also demonstrated that the octopus embryo accumulated 110mAg directly from the dissolved phase and also indirectly through assimilation of the contaminated yolk. These results raise questions regarding the potential contrasting vulnerability of early life stages of cephalopods to the metallic contamination of coastal waters.

Introduction

Cephalopods in European waters have attracted increasing attention from marine biologists and fishery scientists [1]. Indeed, this group is of high ecological significance for the understanding of the food webs in marine ecosystems. This reflects their central position as they are as much a predator of fish and invertebrates as they are prey for numerous fish, seabirds and marine mammals [2,3]. Secondly, cephalopod fisheries have increased significantly over the last six decades, with landings nowadays reaching 120 000 tons per year, i.e., 4-fold the landings made in the 1950s [4]. Moreover, they are prey for numerous fish, seabirds and marine mammals [2,3].

Within the cephalopods, egg encapsulation varies in terms of size, architecture and structure [12]. Briefly, common cuttlefish and sepiolids lay individual medium-sized encapsulated eggs that are covered by gelatinous envelopes stained with ink or by a jelly covered by a rigid capsule, respectively [12]. European squid lay several small-sized eggs that are united by a common ovicapsule and substrates. The eggs therefore develop in an environment that is exposed to more intense anthropogenic activities than adult environments, and to the release of various contaminants including metals. Cephalopod embryos are thus exposed to dissolved trace elements during development, which represents 10–15% of the cephalopod life span [5]. Previous studies of S. officinalis demonstrated that its embryos, which are encapsulated in an eggshell composed of nidamental-, oviducal-originated layers and a chorion, are partially protected against metal penetration [6–8]. Interestingly, these shielding properties varied with respect to the particular trace element, resulting in, for instance, high Ag-accumulation in the embryonic tissues but total protection against Pb-penetration [9–10]. This contrasting metal-dependent permeability of the eggshell can thus limit the exposure of the embryo to the potential toxicological effects of particular contaminants during development, but not others [11].

These coastal cephalopods undergo extensive (e.g., S. officinalis) or limited migration (e.g., O. vulgaris) in the breeding season to reach shallow waters, where they lay their eggs on benthic substrates. The eggs therefore develop in an environment that is exposed to more intense anthropogenic activities than adult environments, and to the release of various contaminants including metals. Cephalopod embryos are thus exposed to dissolved trace elements during development, which represents 10–15% of the cephalopod life span [5]. Previous studies of S. officinalis demonstrated that its embryos, which are encapsulated in an eggshell composed of nidamental-, oviducal-originated layers and a chorion, are partially protected against metal penetration [6–8]. Interestingly, these shielding properties varied with respect to the particular trace element, resulting in, for instance, high Ag-accumulation in the embryonic tissues but total protection against Pb-penetration [9–10]. This contrasting metal-dependent permeability of the eggshell can thus limit the exposure of the embryo to the potential toxicological effects of particular contaminants during development, but not others [11].

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nidelamental jelly sheet, whereas the common octopus produces small eggs that are devoid of an envelope other than the chorion [13]. In this context, this study aimed to determine the potential inter-specific differences in metal bioaccumulation in the embryos and hatchlings of four species (three of which are of high economic interest) that differ appreciably in their eggshell structure. By exposing the eggs to dissolved $^{110m}$Ag, $^{109}$Cd, $^{57}$Co, $^{54}$Mn and $^{65}$Zn throughout development, we determined the concentration and distribution of radiotracers in the eggshell envelopes, vitellus and embryo to help explain the bioaccumulation efficiencies observed in hatchlings of four species, $R$. macronyx, $S$. officinalis, $L$. vulgaris, and $O$. vulgaris, representing the three main Orders of the Class Cephalopoda.

**Materials and Methods**

Adult cuttlefish and octopus, collected by net fishing off Monaco in March 2006 and June 2008, respectively, were maintained in 300-l open-circuit tanks at the IAEA-EL premises and fed daily with crabs and frozen fish. After spawning, cuttlefish eggs ($n = 20$) were collected and immediately separated to optimise their oxygenation. Octopus egg masses (six strings of ~200 eggs each) were kept with the female for brooding during the first week to optimise embryo survival before their collection for experiments. Freshly laid eggs of squid ($n = 6$ strands of ~40 eggs each) and sepilids ($n = 12$ eggs) were collected in May 2008 and March 2009, respectively, off Monaco at a depth of 20–300 m, and placed immediately under experimental conditions.

The eggs were placed in four 20-l glass aquaria containing filtered (0.45 µm) and UV-sterilized seawater (constantly aerated closed circuit; salinity 37 p.s.u.; light/dark cycle: 12 h/12 h) at 16°C, or 19°C for the octopus eggs. They were then exposed to dissolved radiotracers $^{110m}$Ag, $^{109}$Cd, $^{57}$Co (Amersham, UK) and $^{54}$Mn and $^{65}$Zn (Isotope Product Laboratory, USA) throughout development (except octopus eggs, which were exposed one week after spawning), according to the experimental procedure described by Lacour-Labarte et al. [14]. Due to the sensitivity of the radiotracer tool, the low levels of radioactivity used in this work allowed us to monitor trace element accumulation behaviour under environmentally realistic levels of metal exposure [13]. The concentrations of radiotracers in seawater, exposure durations and concentrations of added metals are reported in Table 1.

The eggs were placed at intermediate developmental stages [16] (Table 2) ($n = 6$ and 4 eggs for sepilids and cuttlefish, respectively; $n = 9$ pools of 10–20 eggs from three strands of squid and five pools of 20 eggs for octopus, to obtain sufficient biomass to allow the detection of significant levels of radioactivity) and dissected to determine the distribution of the radiotracers among the different egg compartments, i.e. the different eggshell layers, the embryo, the vitellus and the perivitelline fluid. At hatching, six and four sepiloid and cuttlefish hatchlings, respectively, and nine pools of 10–20 individuals and five pools of 25 individuals of squid and octopus hatchlings, respectively, were weighed and counted.

Radioanalyses were carried out using a high-resolution γ-spectrometry system consisting of four coaxial Germanium (N-or P-type) detectors (EGNC 33-195-R, Canberra® and Eurycis®) connected to a multi-channel analyser and a computer equipped with spectra analysis software (Interwinner® 6). Radioactivities were determined by comparison with standards of known activity and of appropriate geometry. Measurements were corrected for counting efficiency and physical radioactive decay. The counting time was adjusted to obtain a propagated counting error of less than 5% [17].

The accumulation of each radiotracer was expressed as a concentration factor (CF), which is the ratio between radiotracer activity in the egg or egg compartment (Bq g$^{-1}$) and the time-integrated activity in seawater (Bq g$^{-1}$). The results are expressed as mean ± SD.

Whole-body autoradiography was used to reveal the distribution of $^{110m}$Ag in octopus eggs exposed to dissolved $^{110m}$Ag for 1 week. Eggs were embedded in a 2.5% carboxymethylcellulose gel and flash-frozen in a slurry of dry ice in hexane. From these egg pools, 20-µm thick sections were cut with a specially designed cryomicrotome (Leica CM3600), freeze-dried, and placed on phosphor screens (Perkin-Elmer) for 4 to 7 d. After exposure, the screens were scanned with a Cyclone Phosphor Imager (Perkin-Elmer) and $^{110m}$Ag activity in the egg compartments was quantified as previously described [18].

A Kruskal-Wallis test was applied to determine the differences between the CFs calculated for the different egg compartments. All experiments complied with the ethical and welfare considerations relating to the use of cephalopods as experimental animals [19].

**Results and Discussion**

Many studies have shown that cephalopods have the capacity to accumulate trace elements at high levels in their tissues [20–28]. For this reason, the health concerns related to the consumption of cephalopod flesh have been evaluated for different commercial species from different locations [29–33]. In cephalopods, the effects of metals range from direct effects during embryogenesis to various toxic effects on the early juveniles [11,34–36]. Nonetheless, both maternal transfer and dissolved metals can have an impact during the embryo’s development, and may be particularly critical for the success of population recruitment in cephalopods.

This study revealed that metal accumulation (considering all radioisotopes) in embryos varied between the four considered species, such that $\text{CF}_{\text{octopus}}>\text{CF}_{\text{squid}}=\text{CF}_{\text{cuttlefish}}<\text{CF}_{\text{sepilid}}$ (Table 2). Although the maternal transfer of several elements has previously been demonstrated through the vitellus [37], our results are consistent with the natural metal concentrations recorded in hatchlings from the field [38]. This suggests that metal concentrations in hatchlings reflect the environmental conditions.
in which the eggs develop and depend on egg morphology. Once the eggs have been laid, the observed inter-species differences in bioaccumulation therefore result from contrasting waterborne metal uptake that is likely to be due to the type of encapsulation of the embryos, which is specific to the different species. The CF values recorded in embryos at intermediate stages of development (Table 2) suggest differences in the permeabilities of the eggshells towards metals. Indeed, all radiotracers accumulated in the embryonic tissues in squid and octopus whereas only $^{54}$Mn and $^{110m}$Ag, and $^{54}$Mn and $^{65}$Zn, respectively, were detected at significant levels in the internal compartment of sepiolid and cuttlefish eggs.

The selective permeability of the cuttlefish eggshell towards trace elements has previously been interpreted as being linked to the expansion of the chorionic membrane, resulting in egg swelling [9,10,14], which induces the shrinkage and stretching of the envelopes due to the notable increase in egg volume [12]. Nevertheless, the composition of the eggshell, which is rich in carboxyl- and sulphydryl-groups [13], might explain why most metals were associated with this compartment (>90% for $^{109}$Cd, $^{57}$Co, $^{54}$Mn, $^{65}$Zn and >64% for $^{110m}$Ag; results not shown), therefore hindering metal penetration into the internal compartments. More precisely, $^{110m}$Ag, $^{109}$Cd, $^{54}$Mn and $^{65}$Zn were more highly concentrated in the nidamental layers than in the pooled oviducal envelopes and chorion, but $^{57}$Co was found at similar concentrations in these two compartments ($P_{Kruskall-Wallis}$>0.05; Table 2), suggesting that different components of the eggshell provide different levels of shielding according to the metal. The contrasting affinities of metals for the different eggshell parts were more obvious in squid eggs. Thus, the chorion may act as a barrier against $^{109}$Cd and $^{57}$Co penetration as it exhibited CF values 4-10-fold higher (P

\begin{table}[h]
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\begin{tabular}{|l|c|c|c|c|}
\hline
\textbf{Species} & \textbf{R. macrosoma} & \textbf{S. officinalis} & \textbf{L. vulgaris} & \textbf{O. vulgaris} \\
\hline
\textbf{Cont.} & & & & \\
\hline
\textbf{Table 2. Cont.} & & & & \\
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\end{table}
membrane [39] that might favour the retention of metal ions. Such strong shielding properties of the eggshell probably explain the low metal contamination found in the hatchlings despite long-term exposure (i.e., 82 d) to radioisotopes. It is therefore likely that radiotracer uptake occurs rapidly during the last days of embryonic life when the eggcase starts to break before hatching. However, 54Mn was detected in the internal jelly and chorion, suggesting that this essential element can pass selectively through the outer capsule before accumulating in the embryo.

Octopus eggs lack an eggshell and the eggs are only protected by a chorionic membrane, which retained 88, 84 and 72% of the 57Co, 54Mn and 65Zn, respectively (results not shown). However, the retention capacity of the chorion for 110mAg and 109Cd was low (i.e., 4 and <7%, respectively). Consequently, both metals became highly concentrated in the internal compartments (Table 2), resulting in at least 4- and 7-fold higher CF values in the octopus hatchlings than in the other species (Table 1). Our results are consistent with the 5-fold higher Ag concentrations recorded in hatchlings of O. vulgaris compared to those of S. officinalis and L. vulgaris [38]. In cuttlefish, it is noteworthy that the embryo accumulated more of the metals that penetrated through the eggshell than the other compartments (i.e. CFemb>CFyolk>CFPVF; Kruskall-Wallis, P<0.05; Table 2). In contrast, no significant differences between the embryo and the yolk were observed in the CF values of 65Zn in cuttlefish, 57Co, 54Mn, 65Zn in squid and 109Cd in octopus (P>0.05; Table 2). More surprising was that the highest CF values of 109Cd and 110mAg were observed in the yolk of squid and octopus, respectively, compared to the other compartments (P<0.01; Table 2). The autoradiogram of octopus eggs exposed to dissolved 110mAg revealed that metal distribution was closely related to the position of the outer and inner yolk sacs (Figure 1). This clearly indicates that Ag is efficiently incorporated in the vitelline reserves. Thus, the octopus embryo accumulates waterborne Ag both directly from seawater and from the yolk, which absorbs the metal. This would explain the high CF values recorded in octopus paralarvae at hatching (CFAg of 207±3785; Table 2). In this context, the embryos of octopus appeared to be the most highly exposed to environmental contamination by metals, and especially by Ag.

In conclusion, this work shows the strongly contrasting efficiencies of metal uptake by the four studied species, which appears to be correlated with the different embryo encapsulation strategies observed in these cephalopods [12]. The lowest CF values were found in the sepiolid and cuttlefish embryos (despite their longer exposure to radiotracers in this study; see Table 2), both of which are protected by hard shells during their longer embryonic life, unlike those of squid, which are embedded in soft envelopes, or those of octopus that lack any protective eggshell [12]. The large eggs of R. macrosoma and S. officinalis produce large, benthic hatchlings that weigh 30 to 100 times more than the small, planktonic and active swimming hatchlings of L. vulgaris and O. vulgaris [5,40] (Table 2). The different mode of life of these early stages suggests the possibility of a contrasting metabolism between the embryos that could modulate the rate of uptake of metals [41]. In fact, metals were incorporated in inverse relation to body size (i.e. lower for the large-sized S. officinalis and higher for the small-sized O. vulgaris). It can thus be tentatively hypothesized that the surface–volume relationship may influence metal absorption [42,43]. Finally, these results demonstrate different strategies regarding metal requirements and/or protection with respect to the essential and non-essential elements [37].
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