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Authors
Bell, TG
Landwehr, S
Miller, SD
et al.

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Estimation of bubble-mediated air–sea gas exchange from concurrent DMS and CO$_2$ transfer velocities at intermediate–high wind speeds

Thomas G. Bell$^1$, Sebastian Landwehr$^2$, Scott D. Miller$^3$, Warren J. de Bruyn$^4$, Adrian H. Callaghan$^{5,a}$, Brian Scanlon$^2$, Brian Ward$^2$, Mingxi Yang$^1$, and Eric S. Saltzman$^6$

$^1$Plymouth Marine Laboratory, Prospect Place, The Hoe, Plymouth, PL1 3DH, UK
$^2$School of Physics, National University of Ireland, Galway, Ireland
$^3$Atmospheric Sciences Research Center, State University of New York at Albany, NY, USA
$^4$Schmid College of Science and Technology, Chapman University, Orange, California, CA, USA
$^5$Scripps Institution of Oceanography, University of California San Diego, 9500 Gilman Drive, La Jolla, CA 92093, USA
$^6$Department of Earth System Science, University of California, Irvine, CA, USA

$^a$now at: Department of Civil and Environmental Engineering, Imperial College London, South Kensington Campus, London, SW7 2AZ, UK

Correspondence to: Thomas G. Bell (tbe@pml.ac.uk)

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Abstract. Simultaneous air–sea fluxes and concentration differences of dimethylsulfide (DMS) and carbon dioxide (CO$_2$) were measured during a summertime North Atlantic cruise in 2011. This data set reveals significant differences between the gas transfer velocities of these two gases ($\Delta k_w$) over a range of wind speeds up to 21 m s$^{-1}$. These differences occur at and above the approximate wind speed threshold when waves begin breaking. Whitecap fraction (a proxy for bubbles) was also measured and has a positive relationship with $\Delta k_w$, consistent with enhanced bubble-mediated transfer of the less soluble CO$_2$ relative to that of the more soluble DMS. However, the correlation of $\Delta k_w$ with whitecap fraction is no stronger than with wind speed. Models used to estimate bubble-mediated transfer from in situ whitecap fraction underpredict the observations, particularly at intermediate wind speeds. Examining the differences between gas transfer velocities of gases with different solubilities is a useful way to detect the impact of bubble-mediated exchange. More simultaneous gas transfer measurements of different solubility gases across a wide range of oceanic conditions are needed to understand the factors controlling the magnitude and scaling of bubble-mediated gas exchange.

1 Introduction

Air–sea exchange is a significant process for many compounds that have biogeochemical and climatic importance. Approximately 25 % of the carbon dioxide (CO$_2$) released into the atmosphere by anthropogenic activities has been taken up by the world oceans, which has tempered its climate forcing while leading to ocean acidification (Le Quéré et al., 2015). The biogenic gas dimethylsulfide (DMS) is a major contributor to the mass of marine atmospheric aerosol (Virkkula et al., 2006). Volatile organic compounds (VOCs) such as isoprene, acetone and acetaldehyde alter the oxidising capacity of the troposphere (Carpenter et al., 2012). The solubility differences between these VOCs mean that their exchange is controlled to differing degrees by processes on the water and air side of the air–sea interface (Yang et al., 2014). Many of the factors influencing air–sea gas exchange will be altered by future changes in climate, ocean circulation and biology. Earth system models and air quality models require more accurate understanding of the processes that influence air–sea gas transfer.

Air–sea gas exchange is typically parameterised as a function of the ocean–atmosphere bulk concentration difference ($\Delta C$) and the physical mixing induced by wind stress at the...
interface (Liss and Slater, 1974). The air–sea flux is typically described using the expression
\[ \text{Flux} = K(C_w - \alpha C_a), \]
where \( C_w \) and \( C_a \) are the trace gas bulk concentration on either side of the interface, \( \alpha \) is the dimensionless water/air solubility of the gas in seawater and \( K \) is the gas transfer velocity. The physics of gas transfer are implicitly represented by the gas transfer velocity, which is commonly expressed in water-side units of velocity (cm h\(^{-1}\)) and parameterised as a function of wind speed (\( U_{10} \)) and Schmidt number (\( Sc \)).

The simplicity of Eq. (1) belies the complexity of the processes involved in air–sea gas transfer. These processes include diffusion, surface renewal and bubble-mediated transport. In turn, turbulence can be generated by wind stress, wave-induced mixing, buoyancy, currents and wave breaking. A variety of theoretical, laboratory and field approaches have been used to study the processes that control air–sea transfer, but we do not yet have a firm understanding of their relative importance under a range of atmospheric and oceanic conditions.

The gas-transfer–wind-speed relationships for gases of different solubility may be affected by breaking waves and bubbles (Keeling, 1993; Woof, 1993, 1997). Gas invasion and evasion via bubbles (\( k_{\text{bub}} \)) is sensitive to the void fraction (ratio of air volume to total volume) of the bubble plume as well as the bubble size distribution. Bubble injection depth and cleanliness of the surface (influenced by surfactants) affect bubble rise velocity and residence time. Bubble residence time determines the time available for equilibration to occur while bubble volume, pressure and gas diffusivity (\( Sc \)) govern the time needed for a bubble to equilibrate. The magnitude of \( k_{\text{bub}} \) is expected to be greater for sparingly soluble gases (e.g. \( CO_2 \), dimensionless solubility \( \sim 1 \)) than for more soluble gases such as DMS (dimensionless solubility \( \sim 15 \)), particularly when bubbles are fully equilibrated.

Deliberate, dual-tracer techniques have estimated gas transfer by measuring the evasion of a pair of sparingly soluble gases with different diffusivity (\( ^{3}He \) and \( SF_6 \), dimensionless solubility \( \leq 0.01 \)). These studies observed a non-linear wind speed dependence of the gas transfer velocity, in qualitative agreement with earlier studies in wind–wave tanks (e.g. Wanninkhof et al., 1985; Liss and Merlivat, 1986; Watson et al., 1991). Direct, shipboard measurements of water-side gas transfer have also been made by eddy covariance (e.g. McGillis et al., 2001; Huebert et al., 2004; Marandino et al., 2007; Miller et al., 2010; Bell et al., 2013). These measurements typically show DMS gas transfer velocities that are lower and exhibit more linear wind speed dependence than the \( CO_2 \) transfer velocity–wind speed relationship inferred from dual-tracer studies (e.g. Yang et al., 2011; Goddijn-Murphy et al., 2012; Bell et al., 2015). It has been suggested that the difference between the open-ocean gas transfer velocities of \( CO_2 \) and DMS is due to the reduced importance of bubble-mediated exchange for DMS (Blomquist et al., 2006; Fairall et al., 2011; Goddijn-Murphy et al., 2016).

Only one set of concurrent \( CO_2 \) and DMS gas transfer velocity measurements have been published to date (Miller et al., 2009). In that study, no data were collected for winds greater than 10 m s\(^{-1}\) and no statistically significant difference was observed in the \( CO_2 \) and DMS gas transfer–wind speed relationships after normalising both gases to a common diffusivity. This study presents a more extensive set of \( CO_2 \) and DMS gas transfer velocities that were measured simultaneously aboard the R/V Knorr in the 2011 summertime North Atlantic in both oligotrophic and highly productive waters. The DMS gas transfer velocities are discussed separately in detail by Bell et al. (2013). Here we focus specifically on what can be learned about gas transfer from the differences in behaviour of two different solubility gases at intermediate and high wind speeds.

2 Methods

2.1 Seawater, atmospheric and flux measurement systems

The measurement setups for DMS and \( CO_2 \) concentrations in air and water and the eddy covariance flux systems have been discussed in detail elsewhere (Miller et al., 2008; Saltzman et al., 2009; Miller et al., 2010; Bell et al., 2013, 2015; Landwehr et al., 2014; Landwehr et al., 2015). We provide a summary plus some additional details in the Appendix.

2.2 Gas transfer velocity calculations

In this section we describe the calculation of DMS and \( CO_2 \) gas transfer velocities from the Knorr_11 cruise data. Measured gas transfer velocities are transformed into water-side-only gas transfer velocities in order to remove the influence of air-side resistance. The relative contribution of air-side resistance to the total resistance is a function of solubility and thus different for the two gases. Finally, we discuss the most appropriate approach for comparing the water-side gas transfer velocities, given that the two gases have different molecular diffusivity and solubility.

Total gas transfer velocities (\( K \)) are calculated for \( CO_2 \) and DMS for each 10 min flux interval of the Knorr_11 cruise using Eq. (1). The temperature-dependent dimensionless solubilities of \( CO_2 \) and DMS in seawater are calculated following Weiss (1974) and Dacey et al. (1984) respectively. These gas transfer velocities reflect the result of resistance on both sides of the interface (Liss and Slater, 1974). The water-side
contribution to the total resistance is determined as follows:

\[ k_w = \left[ \frac{1}{K' - \frac{\alpha}{k_a}} \right]^{-1} \]

where \( k_w \) and \( k_a \) are the air-side and water-side gas transfer velocities and \( \alpha \) is dimensionless water/air solubility. Note that we use the \( \alpha \) reported by Dacey et al. (1984) in these calculations rather than the Henry law constant \((H, \text{units of atm L mol}^{-1})\) as there appears to be an error in conversion between \( \alpha \) and \( H \) in that study (see Supplement discussion).

CO2 solubility is sufficiently low that air-side resistance is negligible and the water-side gas transfer is assumed equal to the total transfer velocity \((k_{CO2} = K_{CO2})\). The air-side resistance for DMS needs to be accounted for because it is a moderately soluble gas (McGillis et al., 2000). Air-side gas transfer velocities \((k_a)\) for DMS were calculated for each 10 min flux interval with the NOAA COAREG 3.1 model, using sea surface temperature (SST) and horizontal wind speed measured during the cruise. The NOAA COAREG 3.1 model (Fairall et al., 2011) is an extension of the COARE bulk parameterisation for air–sea energy and momentum fluxes to parameterise gas transfer (Fairall et al., 1996, 2000). The air-side resistance contributes about 5% to the total resistance for DMS. NOAA COAREG 3.1 model calculations were carried out using a turbulent/molecular coefficient, \( A = 1.6 \), and bubble-mediated coefficient, \( B = 1.8 \) (Fairall et al., 2011). Knorr_11 measurements of SST, air temperature, relative humidity, air pressure, downward radiation and wind speed were used as input parameters to the model. Note that the use of the COAREG 3.1 model introduces a small uncertainty in our estimates of water-side DMS gas transfer velocity (approximately ±2% when wind speed = 20 m s\(^{-1}\)).

To facilitate comparison of transfer coefficients for the two gases across a range of sea surface temperatures, gas transfer velocities are corrected for changes in molecular diffusivity and viscosity. The correction typically involves the normalisation of water-side gas transfer velocities to a common Schmidt number \((Sc = 660)\), equivalent to CO2 in seawater at 20°C:

\[ k_{X,660} = k_X \cdot \left( \frac{660}{Sc_X} \right)^{-0.5} \]

where subscript \( X \) refers to CO2 or DMS (i.e. \( k_{DMS,660} \) and \( k_{CO2,660} \)). Temperature-dependent \( Sc_{CO2} \) and \( Sc_{DMS} \) were obtained using the in situ seawater temperature from the ship’s bow sensor and parameterisations from Wanninkhof (1992) and Saltzman et al. (1993).

The \( Sc \) number normalisation (Eq. 3) is commonly used across the whole range of wind speeds. In fact, it is only appropriate at low or moderate winds when interfacial gas transfer dominates over bubble-mediated gas exchange. If bubbles are an important component of gas transfer then solubility also plays a role and normalisation based on \( Sc \) alone may not be sufficient.

To develop a more rigorous comparison of \( k_{DMS} \) and \( k_{CO2} \), we normalised the water-side transfer velocities of DMS to the Schmidt number of CO2 at the in situ sea surface temperature of each 10 min flux interval, as follows:

\[ k_{DMS,Sc} = \frac{k_{DMS}}{S_{CO2}/S_{DMS}} \]

where \( S_{CO2} \) and \( S_{DMS} \) are the Schmidt numbers of CO2 and DMS at the in situ sea surface temperature. Compared to normalising both DMS and CO2 to \( Sc = 660 \), this approach has the advantage of correcting only \( k_{DMS} \), with no correction to \( k_{CO2} \). The Sc correction for DMS should be reasonably accurate, assuming that the bubble-mediated transfer for the more soluble DMS is relatively small.

On the Knorr_11 cruise, the variability in sea surface temperature was small (1σ = ±1°C). As a result, there is little difference in the variability or wind speed dependence of \( Sc \)-corrected \( k_{CO2} \) compared to \( k_{CO2} \) at the in situ temperature (Fig. 5 vs. Fig. S5 in Supplement). In Sect. 3.4, the relationship between CO2 and DMS gas transfer velocities and wind speed is examined using \( k_{DMS,Sc} \) and \( k_{CO2} \).

### 2.3 Calculation of \( k_{bub,CO2} \)

The flux of a water-side controlled gas is equal to the sum of the interfacial flux and the bubble-mediated flux. For gases with significant air–sea disequilibrium these processes are often considered as parallel transfer velocities, i.e. total transfer velocity \( k_w = k_{int} + k_{bub} \). See Woolf (1997) for a more complete discussion of bubble-mediated transfer for gases close to ocean–atmosphere equilibrium. We assume that turbulence and diffusive mixing at the sea surface operate similarly upon the interfacial air–sea transfer of CO2 and DMS (i.e. \( k_{int,CO2} = k_{int,DMS} \)), given appropriate normalisation for the differences in molecular diffusivity. Observed differences between \( k_{DMS,Sc} \) and \( k_{CO2} \) should therefore be a measure of the difference between the bubble-mediated contributions to DMS and CO2 gas transfer:

\[ \Delta k_w = k_{bub,CO2} - k_{bub,DMS} \]

Strictly speaking, Eq. (5) should also account for the influence of bubble overpressure, which alters the gas flux due to bubbles when the concentration gradient is into the ocean. The extra pressure on the gas in the bubbles is calculated following Woolf (1997): \( \Delta = (U_{10}/U_i)^2 \) where \( U_i \) is the wind speed at which the supersaturation of a particular gas equals 1% (49 m s\(^{-1}\) in the case of CO2). A high wind speed (20 m s\(^{-1}\)) gives \( \Delta = 0.167 \), which would lead to only a ~2% enhancement of the CO2 flux when the air–sea concentration gradient is 30 ppm (minimum for this study) and into the ocean. The magnitude of this effect would be larger for gases less soluble than CO2 but we are able to ignore it for the purposes of this study.

\( k_{bub,CO2} \) and \( k_{bub,DMS} \) are related by the influence of solubility and diffusivity upon bubble-mediated transfer. We pa-
rameterise this relationship simply as $k_{\text{bub, DMS}} = f k_{\text{bub, CO}_2}$. Substitution into Eq. (5) yields

$$k_{\text{bub, CO}_2} = \frac{\Delta k_w}{1 - f}.$$  

(6)

The value of $f$ depends on seawater temperature and the complex dynamics of bubble formation and cycling (size distributions, surfactants, etc.). At the mean SST encountered in this study (9.8°C), the bubble gas transfer models of Woolf (1997) and Asher (Asher and Wanninkhof, 1998; Asher et al., 2002) yield values for $f$ of 0.14 and 0.28, respectively (see Supplement for model equations).

### 2.4 Sea surface imaging

Whitecap areal fraction was measured using images of the sea surface recorded with a digital camera (5-megapixel Arc-ent Vision, 16 mm focal length lens) mounted 14.6 m above the ocean surface at an angle of ~75° from the nadir. Image footprints represent ~7600 m² of sea surface. Images were collected at a sample interval of about 1 s and post-processed for whitecap fraction according to the automated whitecap extraction algorithm method (Callaghan and White, 2009). More detail on the methodology, camera exposure settings and data comparability are provided in the Supplement. Images were further processed to distinguish whitecap pixels as either stage A or stage B whitecaps by applying a spatial separation technique (Scanlon and Ward, 2013). The whitecap fraction measurements were averaged in the same way as the gas transfer velocities (i.e. time-averaged mean values as well as 2 m s⁻¹ wind speed bins).

### 3 Results

#### 3.1 Cruise location and environmental conditions

This study took place in the summertime North Atlantic (24 June–18 July 2011; DOY 175–199), departing and returning to Woods Hole, MA. Most of the data were collected north of 50° N, including the occupation of four 24–36 h stations – ST181, ST184, ST187 and ST191 (Fig. 1). The cruise track was designed to sample regions with high biological productivity and phytoplankton blooms, with large air–sea concentration differences for CO₂ and DMS. The cruise meteorology and physical oceanography is discussed in detail by Bell et al. (2013). A series of weather systems travelling from West to East passed over the region during the cruise. Wind speeds ranged from ~ 1 to 22 m s⁻¹, with strongest winds during the frontal passages at stations ST184 and ST191 (Fig. 1b). Atmospheric boundary layer stability was close to neutral for most of the cruise ($z/L < 0.07; 75\%$ of the time), with infrequent stable conditions ($z/L > 0.05; < 8\%$ of the time, Fig. 1a). There was no evidence that the stable periods affected the flux measurements (Bell et al., 2013). Whitecap areal fraction increased up to a maximum of ~0.06 in response to high wind speeds (Fig. 1b).

#### 3.2 Whitecaps

Whitecaps were observed during Knorr_11 when wind speeds exceeded 4.5 m s⁻¹, a typical wind speed threshold for whitecap formation in the open ocean (Callaghan et al., 2008; Schwendeman and Thomson, 2015). Whitecap areal fraction is a strong, non-linear function of wind speed (Fig. 2a). The whitecap vs. wind speed relationship for Knorr_11 is similar in shape to recently published wind-speed-based whitecap parameterisations (Callaghan et al., 2008; Schwendeman and Thomson, 2015). At intermediate wind speeds the Knorr_11 whitecap data are lower than the parameterisations (Fig. 2a). Total whitecap coverage is a function of (i) active “stage A whitecaps” ($W_A$) produced from recent wave breaking and (ii) maturing “stage B whitecaps” ($W_B$) that are decaying foam from previous breakers. The Stage A whitecap fraction data are highly variable at ~11 m s⁻¹ wind speeds (Fig. 2b), which is driven by the difference in the wind–wave conditions during Knorr_11 (see discussion in Supplement).

#### 3.3 Concentrations, fluxes and gas transfer velocities

Seawater $\rho_{\text{CO}_2}$ was consistently lower than the overlying atmosphere throughout the study region due to biological uptake (Fig. 3a). As a result, the air–sea concentration difference ($\Delta \rho_{\text{CO}_2}$) was large and always into the ocean, with $\Delta \rho_{\text{CO}_2} < -45$ ppm for more than 80% of the measurements. Periods with particularly enhanced $\Delta \rho_{\text{CO}_2}$ into the ocean were during the transit between ST181 and ST184 ($\Delta \rho_{\text{CO}_2}$ as large as ~120 ppm) and during ST191 ($\Delta \rho_{\text{CO}_2}$ consistently ~75 ppm).

Seawater DMS levels were much higher than atmospheric levels, reflecting the biogenic sources in seawater and the relatively short atmospheric lifetime (~1 day; Kloster et al., 2006). The largest air–sea DMS concentration differences ($\Delta \text{DMS}$) of 6–12 ppb were observed during DOY 185–190 (Fig. 4a). The $\Delta \text{DMS}$ and $\Delta \rho_{\text{CO}_2}$ did not co-vary (Spearman $\rho = 0.11, n = 918, p < 0.001$). This is not surprising because, although seawater DMS and CO₂ signals are both influenced by biological activity, they are controlled by different processes. Seawater CO₂ levels reflect the net result of community photosynthesis and respiration, while DMS production is related to metabolic processes that are highly species-dependent (Stefels et al., 2007).

$\text{CO}_2$ fluxes ($F_{\text{CO}_2}$) were generally into the ocean, as expected given the direction of the air–sea concentration difference (Fig. 3b). The variability in $F_{\text{CO}_2}$ observed on this cruise reflects dependence on both wind speed and $\Delta \rho_{\text{CO}_2}$. For example, during DOY182 air-to-sea CO₂ fluxes increase due to a gradual increase in $\Delta \rho_{\text{CO}_2}$ with fairly constant wind speed. More commonly, $\Delta \rho_{\text{CO}_2}$ was fairly constant and...
Figure 1. Time series of 10 min averaged data collected during the Knorr_11 cruise. Dashed black line in panel (a) indicates neutral atmospheric stability ($z/L = 0$). Grey shaded regions represent intervals when the ship occupied stations ST181, ST184, ST187, and ST191. Measured wave properties (see Bell et al., 2013) are presented in panels (c) and (d): significant wave height $H_s$ (c) and inverse wave age (d). $U_{10n}/C_p \geq 1$ represent younger seas and $U_{10n}/C_p < 1$ represent older seas.

Figure 2. Semi-log plots of whitecap areal fraction as a function of mean horizontal wind speed at 10 m above the sea surface ($U_{10n}$) during the Knorr_11 cruise. 10 min average (grey dots) and 2 h average (red triangles) data are shown on both panels. (a) Total whitecap area ($W_T$) versus $U_{10n}$ bin-averaged data (open squares, 2 m s$^{-1}$ bins). The best fit line to Knorr_11 2 h average data (green; $\log_{10} (W_T) = -42.19 e^{(-0.95U)} - 6.5e^{(-0.0886U)}$) and wind speed parameterisations from the recent literature are shown for reference. (b) Whitecap area considered to be solely from wave breaking (Stage A whitecaps ($W_A$); see text for definition).
variability in $F_{CO_2}$ reflected changes in wind speed. For example, from DOY 185 to 187 wind speeds gradually declined from $\sim 10$ to $5 \text{ m s}^{-1}$ with a concurrent decline in $F_{CO_2}$. DMS eddy covariance fluxes were always out of the ocean. The 10 min averaged DMS fluxes ($F_{DMS}$) clearly show the influence of $\Delta DMS$ (e.g. DOY 188) and wind speed (e.g. DOY 184).

Gas transfer velocities of CO$_2$ and DMS from this cruise exhibit two systematic differences: (i) $k_{DMS}$ values are generally lower than $k_{CO_2}$, particularly during episodes of high wind speed; and (ii) $k_{CO_2}$ is characterised by much larger scatter than $k_{DMS}$. We attribute the large scatter in $k_{CO_2}$ to the greater random uncertainty associated with the eddy covariance measurement of air–sea CO$_2$ fluxes compared to those of DMS. As shown by Miller et al. (2010), the analytical approach used in this study (dried air, closed path LI7500) has sufficient precision to adequately resolve the turbulent fluctuations in atmospheric CO$_2$ associated with the surface flux over most of the cruise ($\Delta p_{CO_2} < \sim 30 \text{ ppm}$). The scatter in the CO$_2$ flux measurements is more likely due to environmental variability resulting from fluctuations in boundary layer CO$_2$ mixing ratio arising from horizontal and/or vertical transport unrelated to air–sea flux (Edson et al., 2008; Blomquist et al., 2014). These effects likely have a much smaller effect on air–sea DMS fluxes, because the air–sea DMS concentration difference is always much larger than the mean atmospheric DMS concentration (due to the short atmospheric lifetime of DMS). For example, a $\Delta p_{CO_2}$ of 100 ppm at a wind speed of $10 \text{ m s}^{-1}$ will produce turbulent fluctuations that are $\sim 0.02 \%$ of the background CO$_2$ on average. In contrast, a typical seawater DMS concentration (2.6 nM) at a wind speed of $6 \text{ m s}^{-1}$ generates fluctuations that are $20 \%$ of the background (Table 1; Blomquist et al., 2012). Thus, $F_{CO_2}$ measurements are highly sensitive to small fluctuations in background CO$_2$ and the relative uncertainty is expected to be much larger than that for $F_{DMS}$.

### 3.4 Comparison of $k_{CO_2}$ and $k_{DMS,Sc}$

The differences between CO$_2$ and DMS gas transfer velocities observed in the time series are also evident when the data are examined as a function of wind speed. From the 10 min averaged data, it is clear that $k_{CO_2}$ is greater than $k_{DMS}$ and has a stronger wind speed dependence over most of the wind speed range (Fig. 5a, b). These broad trends are also easily seen in longer time-averaged data. Flux and $\Delta C$ measure-
Figure 4. Knorr_11 cruise time series of 10 min averaged DMS: (a) air–sea concentration difference (ΔDMS), (b) flux (F_{DMS}) and (c) gas transfer velocity normalised to the in situ CO$_2$ Sc number (k_{DMS,Sc}). Panel (c) also shows k_{DMS,Sc} calculated using NOAA COARE model output (black line). Grey shaded regions represent periods on station.

DMS gas transfer velocities on this cruise exhibit complex behaviour at intermediate to high wind speeds, as discussed in Bell et al. (2013). k_{DMS,Sc} increases linearly with wind speed up to \sim 11 m s$^{-1}$ (Fig. 5). Under the sustained high-wind, high-wave conditions encountered during ST191, the wind speed dependence of k_{DMS,Sc} was lower than expected, with a slope roughly half that of the rest of the cruise data. This effect was not observed at ST184 – for detailed discussion, see Bell et al. (2013). Such coherent spatial-temporal variation means that wind speed bin averaging of the higher wind speed k_{DMS,Sc} may mask real variability in the relationship with wind speed. Relationships developed from wind speed bin-averaged gas transfer data should be interpreted with caution, especially when it comes to developing generalisable air–sea gas transfer models.

The Knorr_11 k_{CO_2} data also demonstrate a clear wind speed dependence (Fig. 5). The NOAA COARE model for CO$_2$ has been tuned to previous eddy covariance flux measurements (McGillis et al., 2001), with bubble-mediated transfer determining the non-linear relationship with wind speed (Fairall et al., 2011). There is reasonable agreement between the COARE model gas transfer velocity predictions and the Knorr_11 k_{CO_2} data up to \sim 11 m s$^{-1}$ wind speed. Above 11 m s$^{-1}$, the COARE model overpredicts k_{CO_2}. This could be interpreted as indicating high wind speed suppression of gas transfer for CO$_2$ as observed for DMS (as discussed by Bell et al., 2013). However, it is important to note that the number of high wind speed (> 15 m s$^{-1}$) gas transfer measurements in this study is limited to 9 and 16 h of data for DMS and CO2 respectively. Much more data are needed in order to firmly establish the high wind speed behaviour.

The COAREG 3.1 model parameterises interfacial gas transfer by scaling to Sc and friction velocity and estimates bubble-mediated gas transfer following Woolf (1997). The lower solubility of CO$_2$ leads to enhanced gas transfer relative to that of DMS at high wind speeds where bubble transport is significant (Fairall et al., 2011). There is good
agreement between the COAREG model gas transfer velocity predictions and the Knorr_11 $k_{CO_2}$ and $k_{DMS}$ data until \( \sim 11 \, \text{m s}^{-1} \) wind speed.

Earlier in this paper we introduced the quantity $\Delta k_w$ as an observational measure of the difference in gas transfer velocities of CO$_2$ and DMS (Sect. 2.3, Eq. 6). The relationship between $\Delta k_w$ and wind speed is positive and shows no systematic differences related to temporal variability (Fig. 6). Sea surface temperature (SST) is indicated by symbol size. Some of the scatter in Fig. 6 could be driven by changes in Sc due to SST variability. Nearly all of the data in Fig. 6 are from periods when SST was relatively constant (9.7 ± 1.1°C). Many of the $k_{CO_2}$ data with warm seawater (i.e. ST181, SST > 12°C) were rejected by our quality control criteria (see Appendix A3). These data were collected when wind speeds were low, which resulted in small CO$_2$ fluxes with large variability at low frequencies. Of the periods with SST > 12°C that passed the quality control criteria, the majority contributed fewer data within a 2 h averaging period than the minimum threshold (three 10 min averaged data points).

4 Discussion

The bubble-mediated component of gas transfer is a strong function of wind speed and breaking waves. Previous estimates of bubble-mediated air–sea gas exchange have used data from laboratory experiments (Keeling, 1993; Asher et al., 1996; Woolf, 1997). The differences between gas transfer velocities for DMS and CO$_2$ provide a unique way to constrain the importance of bubble-mediated transfer under
natural conditions. This study shows that $\Delta k_w$ is near zero ($<4.5 \text{ cm h}^{-1}$) at low wind speeds ($U_{10} \leq 4.5 \text{ m s}^{-1}$), which is consistent with the wind speed at which whitecap fraction becomes significant ($W_T > 10^{-3}$, Fig. 2a). Above 4.5 m s$^{-1}$, $\Delta k_w$ increases non-linearly, consistent with an increase in bubble-mediated CO$_2$ transfer associated with wave breaking. The relationship between $\Delta k_w$ and wind speed is non-linear, and a power law wind speed dependence yields a good fit ($R^2 = 0.66$; Fig. 6):

$$\Delta k_w = 0.177U_{10}^{1.928}. $$

(7)

The functional form of this relationship is qualitatively consistent with those found between $U_{10}$ and breaking waves/wave energy dissipation (Melville and Matusov, 2002) and $U_{10}$ vs. whitecap areal fraction (e.g. Callaghan et al., 2008; Schwendeman and Thomson, 2015). Bubble-mediated gas transfer is the only viable explanation for the magnitude and wind speed dependence of $\Delta k_w$. The only alternative explanation would require a large systematic bias in the measurement of relative gas transfer velocities of DMS and CO$_2$. There are no obvious candidates for such biases.

During strong wind/large wave conditions, the Knorr_11 data suggest that bubble-mediated exchange is a dominant contributor to the total transfer of CO$_2$. For example, when wind speeds were 11–12 m s$^{-1}$, $\Delta k_w$ was about 50% of the total CO$_2$ gas transfer ($k_{CO_2}$). A significant contribution by bubbles to the total gas transfer velocity means that bubble-mediated exchange must be included and adequately parameterised by gas transfer models. The Schmidt number ($Sc$) normalisation (Eq. 4) assumes that the gas transfer velocity is purely interfacial. An alternative normalisation (involving $Sc$ and solubility) is required when bubble-mediated transfer is significant. Our data suggest that the current $Sc$ normalisation should be applied with caution to gas transfer data for different solubility gases at wind speeds greater than 10 m s$^{-1}$.

If $\Delta k_w$ reflects the difference between the bubble-mediated contribution to the transfer of CO$_2$ and DMS, one would expect $\Delta k_w$ to correlate with wave-breaking, and hence with the areal coverage of whitecaps. Breaking waves generate plumes of bubbles (Stage A whitecaps, $W_A$), which then rise to the surface and persist for a short period as foam (Stage B whitecaps, $W_B$). Almost all whitecap measurements represent the fraction of the sea surface that is covered by bubble plumes and/or foam – i.e. $W_T = W_A + W_B$. The $\Delta k_w$ is positively correlated with both $W_T$ (Spearman $\rho = 0.65$, $n = 43$, $p < 0.001$) and $W_A$ (Spearman $\rho = 0.74$, $n = 32$, $p < 0.001$; Fig. 7a, b). These correlations are approximately the same strength as the correlation between $\Delta k_w$ and wind speed (Spearman $\rho = 0.73$, $n = 88$, $p < 0.001$). The functional form of the relationship between $\Delta k_w$ and whitecap areal extent appears to be linear for $W_T > 0.005$. However, the Knorr_11 data set is small and quite scattered, particularly when $W_T < 0.005$. More data are required to fully test the validity of whitecap areal fraction as a proxy for bubbles and bubble-mediated exchange.

Observations of the decaying white cap signal ($W_B$) suggest that the persistence of surface foam is related to both bubble plume depth (deeper bubble plumes take longer to degas) and sea surface chemistry (Callaghan et al., 2013). As measured here, $W_B$ is approximately an order of magnitude larger than $W_A$ and thus dominates the $W_T$ signal. It is often assumed that gas exchange takes place in bubble plumes formed by active wave breaking (i.e. $W_A$), while $W_B$ may vary widely due to surfactant concentration with little or no impact upon bubble-mediated gas exchange (e.g. Pereira et al., 2016). In this case, $\Delta k_w$ should be more strongly correlated with $W_A$ than $W_B$ or $W_T$. The Knorr_11 data do not suggest that $W_A$ is an improvement upon either $W_T$ or even wind speed as a measure of bubble-mediated exchange. This may be because whitecaps do not fully represent the bubbles facilitating gas exchange as these may dissolve before...
they reach the sea surface. Alternatively, $W_T$ and $W_A$ may be equally good (or poor) proxies for bubbles because: (i) surfactant activity was either insignificant or sufficiently invariant in the study region (despite high biological productivity) that $W_B$ does not confound the relationship between $W_T$ and $W_A$; (ii) $W_A$ is no better than $W_T$ at representing the volume of air entrained by breaking waves; and/or (iii) bubbles residing at the surface (i.e. $W_B$) continue to contribute to gas transfer (Goddijn-Murphy et al., 2016).

As shown earlier, the bubble-mediated contribution to gas transfer ($k_{bub, CO_2}$) can be obtained from $\Delta k_w$ using information from mechanistic bubble gas transfer models ($f$; see Sect. 2.3). The $k_{bub, CO_2}$ data sets derived from the Knorr_11 data using the Asher (Asher and Wanninkhof, 1998; Asher et al., 2002) and Woolf (1997) models differ by about 15% (Fig. 8). The field-based estimates of $k_{bub, CO_2}$ can also be compared to model-only estimates for the Knorr_11 conditions using the Asher and Woolf models. Both models are based on total whitecap areal fraction, $W_T$. A non-linear fit of the Knorr_11 $W_T$ and wind speed measurements ($W_T = 1.9 \times 10^{-6} U_10^{-0.36}$) was used to drive both models (Fig. 8). The Asher et al. (2002) model is based on laboratory tipping bucket gas evasion experiments (Asher et al., 1996) and the model was then adjusted to represent the flux of CO$_2$ into the ocean (invasion). Woolf (1997) scaled a single bubble model to the open ocean based on laboratory experiments.

Both models significantly underestimate $k_{bub, CO_2}$ at wind speeds below about 11 m s$^{-1}$. At higher wind speeds, the Asher et al. (2002) model increases rapidly with wind speed to agree better with the Knorr_11 data. In contrast, Woolf (1997) consistently underestimates $k_{bub, CO_2}$ at all wind speeds. Both $k_{bub, CO_2}$ models depend on the choice of wind speed versus whitecap parameterisation. Using the Schwendeman and Thomson (2015) whitecap parameterisation instead of the Knorr_11 best fit makes some difference to the model output, but not enough to adequately fit to the data (Fig. 8). A “dense plume model” was also developed by Woolf et al. (2007) to take account of the interaction of a bubble plume with the interstitial water between bubbles. This model yields estimates of $k_{bub, CO_2}$ that are even lower than the original Woolf (1997) “single bubble model” (data not shown).

It is likely that the Knorr_11 cruise data will be compared with estimates of $k_{bub, CO_2}$ derived from future field campaigns, which will be conducted under different environmental conditions. Our $k_{bub, CO_2}$ data are at in situ seawater temperature ($\sim 10^\circ$C) and thus in situ CO$_2$ solubility ($\alpha = 1.03$) and diffusivity ($Sc = 1150$). We use the Asher et al. (2002) and Woolf (1997) bubble models to make estimates of $k_{bub, CO_2}$ normalised to a standard seawater temperature of $20^\circ$C ($k_{bub, CO_2, 20^\circ}$, where $\alpha = 0.78$ and $Sc = 666$). The 2h averaged Knorr_11 cruise data, including estimates of $\Delta k_w$, $k_{bub, CO_2}$ and $k_{bub, CO_2, 20^\circ}$, are provided in Supplemental Table S1.

The approach used in this study to estimate $\Delta k_w$ and $k_{bub, CO_2}$ from the Knorr_11 field data neglects the effect of sea surface skin temperature and CO$_2$ chemical enhancement. Skin temperature is typically only a few tenths of a degree less than bulk seawater under the conditions encountered in this study (Fairall et al., 1996). The impact upon $k_{CO_2}$ due to skin temperature effects on CO$_2$ solubility and carbonate speciation is probably of the order of 3% (Woolf et al., 2016); there is a chemical enhancement of the CO$_2$ flux due to ionisation at the sea surface (Hoover and Berkshire, 1969). The effect on $k_{CO_2}$ has been estimated to be up to about 8% at a wind speed of 4–6 m s$^{-1}$ (Wanninkhof and Knox, 1996), which amounts to a maximum impact of a few cm h$^{-1}$. By neglecting these effects we have slightly overestimated $\Delta k_w$ and $k_{bub, CO_2}$, but the magnitude of these corrections would be small relative to the environmental scatter or measurement uncertainty.

5 Conclusions

The Knorr_11 concurrent measurements of DMS and CO$_2$ gas transfer velocities show significant differences in gas transfer between the two gases at intermediate–high wind speeds. These data indicate that (i) bubble-mediated gas transfer becomes significant for CO$_2$ at or above the threshold for wave-breaking and (ii) the wind speed dependence is...
non-linear, with a similar functional form to proposed relationships predicting whitecap areal extent from wind speed. However, existing models of bubble-mediated gas transfer using the Knorr_11 in situ observations of whitecap fraction significantly underestimate the importance of this process.

There are a number of assumptions behind model estimates of bubble-mediated gas exchange (Goddijn-Murphy et al., 2016). Model bias can be crudely split into (i) uncertainties in the scaling of whitecap fraction to the bubble population (e.g. using Cipriano and Blanchard, 1981) and (ii) the relationship between gas exchange and bubble properties, which are predicted as a function of air entrainment into the surface ocean by a breaking wave, bubble injection depth, size distribution and mobility through the water (a function of surface cleanliness and surfactants). The underestimation of bubble-mediated CO$_2$ gas transfer by both models is particularly apparent at low–intermediate wind speeds and low whitecap fraction. This could indicate either that bubble production during microscale breaking is an important process for gas transfer or that the relationship between whitecap fraction and bubble population is poorly constrained.

In summary, the approach of using simultaneous measurements of multiple gases with different solubility appears to be a viable way to constrain the magnitude of bubble-mediated gas transfer. Analysis of additional sparingly soluble gases, such as methane or oxygenated hydrocarbons, would further strengthen this approach. A much larger data set, under a wider range of oceanographic conditions, is certainly needed. In particular, it would be useful to examine DMS and CO$_2$ gas transfer velocities in ocean regions with different temperatures, where the solubility of each gas is significantly different from this study.

Data availability. Relevant data for this paper can be found in Table S1 of the Supplement.
Appendix A

A1 Seawater CO\textsubscript{2} and DMS measurements

Seawater CO\textsubscript{2} and DMS were monitored in the supply of seawater pumped continuously through the ship from an intake on the bow located 6 m below the sea surface. CO\textsubscript{2} was equilibrated with air in a recirculating showerhead type system. Alternate air- and water-side pCO\textsubscript{2} were each measured for 5 min by the same infrared gas analyser (IRGA). Seawater DMS was equilibrated with DMS-free air in a tubular porous membrane equilibrator, operated in a single-pass, counterflow mode. DMS was measured at 1 Hz using chemical ionisation mass spectrometry and bin-averaged at 1 min intervals (UCI miniCIMS; Saltzman et al., 2009). DMS was calibrated by continuously pumping an internal standard of tri-deuterated, DMS (d\textsubscript{3}-DMS) into the seawater flow just before the equilibrator. Details of the methods and instrumentation used for equilibration and detection of seawater DMS are described in Saltzman et al. (2009).

A2 Mast-mounted instrumentation and data acquisition

The eddy covariance system was mounted 13.6 m above the sea surface on the bow mast. Platform angular rates and accelerations were measured by two Systron Donner Motion Pak II (MPII) units. Three-dimensional winds and sonic temperature were measured by two Campbell CSAT3 sonic anemometers. Air sampling inlets for DMS and CO\textsubscript{2} were located at the same height as the anemometers and within 20 cm of the measurement region. GPS and digital compass output were digitally logged at 1 Hz. Winds were corrected for ship motion and orientation as described in Miller et al. (2008) and Landwehr et al. (2015). The eddy covariance data streams were logged in both analogue and digital format as described in Bell et al. (2013).

A3 High-frequency atmospheric DMS and CO\textsubscript{2} measurements

Atmospheric DMS measurements were made at 10 Hz using an atmospheric pressure chemical ionisation mass spectrometer located in a lab van (UCI mesoCIMS; Bell et al., 2013). Air was drawn to the instrument through a 28 m long 1/2 in OD Teflon tube. A subsample of the air stream was passed through a Nafion drier prior to entering the mass spectrometer. The measurement was calibrated using an internal gas standard of tri-deuterated DMS added to the inlet (see Bell et al., 2013).

Atmospheric CO\textsubscript{2} measurements were made on air drawn at 8 L min\textsuperscript{-1} through a filtered inlet (90 mm diameter with 1 micron pore size, Savillex) near the sonic anemometers on the bow mast, through 5 m of 5.9 mm ID polyethylene-lined Dekabon tubing to two fast-response CO\textsubscript{2}/H\textsubscript{2}O IRGAs in an enclosure on the bow mast. The IRGAs were open-path-style sensors (LI7500, Licor Inc.) converted to a closed-path configuration (see Miller et al., 2010) and were plumbed in series. A Nafion multi-tube membrane drier (PD-200T, PermaPure) with 6 L min\textsuperscript{-1} dry air counter flow was installed between the two IRGAs such that the upstream IRGA sampled undried air and the downstream IRGA sampled the same air after drying. This technique removes 97% of the Webb Correction from the measured CO\textsubscript{2} flux (first shown by Miller et al., 2010, and confirmed by Landwehr et al., 2014). The air flow through both the CO\textsubscript{2} and DMS inlets was fully turbulent (Re > 10\textsuperscript{4}). The inlets used in this study introduced a small delay (\(\Delta t = 2.2\) s for DMS, \(\Delta t = 1.2\) s for CO\textsubscript{2}) between measured wind and atmospheric measurements, as well as minor loss of covariance at high frequencies (< 5%). The methods used to estimate the delay and loss of flux are given in Bell et al. (2013).

Eddy covariance fluxes were computed for DMS and CO\textsubscript{2} as \(F_{\text{DMS}}\) or \(F_{\text{CO}_2} = \sigma_{\text{air}} (w'c')\), where \(\sigma_{\text{air}}\) is the dry air density, \(w'\) is the fluctuation in vertical winds and \(c'\) is the delay-adjusted fluctuation in gas concentration. Average covariance fluxes were processed in 10 and 9.5 min intervals for DMS and CO\textsubscript{2}, respectively (hereafter referred to as 10 min intervals). Momentum and sensible heat fluxes were also computed for 10 min intervals (see Bell et al., 2013).

Sampling intervals with a mean wind direction relative to the bow of >90\(^\circ\) were excluded from the final data set. CO\textsubscript{2} fluxes were also excluded from intervals when either (i) relative wind direction changed excessively (SD > 10\(^\circ\)), (ii) relative wind speed was low (< 1 m s\textsuperscript{-1}) or (iii) \(\Delta CO_2\) was low (< 30 ppm). DMS and CO\textsubscript{2} fluxes were quality controlled for excessive low-frequency flux as described in the Supplement of Bell et al. (2013). These quality control criteria excluded 62% of the intervals for CO\textsubscript{2} and 55% for DMS and significantly reduced the scatter in the data.
The Supplement related to this article is available online at https://doi.org/10.5194/acp-17-9019-2017-supplement.

Competing interests. The authors declare that they have no conflict of interest.

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