Smectite appearance in the footwall of the Úrkút manganese ore deposit, Bakony Mts., Hungary

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The Úrkút manganese ore deposit (Transdanubian Range, Hungary) is one of the largest manganese accumulations to be formed during the Toarcian Oceanic Anoxic Event. In the past 60 years, the area was investigated intensively. The core storage facility of the manganese mine had more than 20,000 sample pieces. Most of these samples have never been investigated. During this study, which is the first widespread clay mineral study in the footwall of the Úrkút manganese ore deposit, we investigated 40 samples from seven boreholes (footwall rocks, black/gray shales below and above the first ore bed, and manganese carbonate ores). Although previous studies assumed that smectite is associated only with the ore beds, our research revealed its appearance in the footwall (Pliensbachian) as well. Simultaneously, tripoli (the local name of completely bleached chert) can also be found in the footwall. Based on the investigated samples, a sharp geochemical difference was detected between Pliensbachian and Toarcian sediments. In this paper, we try to trace the relationship between the smectite content of the footwall and the ore bed and compare these results with the observed geochemical changes. Based on the new data, we assume that the ore accumulation was caused by a flow system (upwelling-controlled ore formation).

Keywords: Toarcian Oceanic Anoxic Event, Hungary, Transdanubian Range, Úrkút, manganese deposit, smectite

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Introduction

The sediment-hosted Úrkút manganese ore deposit is the largest manganese accumulation in Hungary. Mining activity lasted almost 100 years (1917–2017). The area was continuously under scientific investigation: several generations of researchers have been working there. After the end of production, there are still unexplained geologic details, which show the complexity of the deposit.

In a project between 2014 and 2016 (13 weeks during the summers), the full core storage facility of the mine was rearchived: more than 20,000 pieces of samples were photographed, repacked, and listed. The work was carried out by some 50 students from the University of Miskolc, Eötvös Loránd University, and the University of Szeged. During this work, many unique and rare samples were found, which could lead to a new wave of investigation. The main goal of our project is to understand the environmental changes that caused the ore accumulation.

Leskó (2014) found smectite below the Úrkút manganese ore deposit in the footwall. The core of our ongoing research is an extended high-resolution, sampling-based investigation, which focuses on the Pliensbachian/Toarcian boundary. We expressed our intentions to understand the environmental changes and the processes in the transition between the Pliensbachian and Toarcian. The aim of this paper is to summarize the initial investigations, which deal with the smectite content in the footwall rocks of the Úrkút deposit, and to explain the appearance of smectite below the ore deposit. Since clay minerals are very sensitive to their environment and its changes, we carried out a study of clay minerals.

Geologic setting

The Úrkút manganese ore deposit is located in the Bakony Mts., close to Lake Balaton, in the Transdanubian Range (TDR) of Western Hungary (Fig. 1). The TDR belongs to the ALps–CArpathians–PAnnonian Basin Mega unit (Csontos et al. 1992). The original location of the TDR was somewhere between the South Alpine and Upper Austroalpine lithosphere fragments (Kázmér and Kovács 1985; Haas et al. 1995; Gawlick et al. 1999; Csontos and Vörös 2004). The geologic succession of the TDR begins in the early Paleozoic with marine sediments, which underwent low-grade metamorphism. It continues with eroded sediments of the Variscan complex (Majoros 1980; Árkai et al. 1987; Fülöp 1990; Haas et al. 1995). Following Variscan uplifting and erosion (Late Carboniferous – Early Permian), regional subsidence resulted in thick continental sediments. From the Middle–Late Permian, the sediment accumulation was influenced by a regional transgression. The sea-level rise changed the type of sedimentation (Late Permian): above the siliceous continental sedimentary rocks, siliceous-carbonate rocks (marls) occur; later, the terrigenous input ceased and carbonate platforms began to be built up. By the late Triassic an extensive, coherent, few km-thick Dachstein-type platforms had developed (Haas 1988; Haas and Budai 1995;
Fig. 1
Location of the Úrkút manganese ore deposit (after Vörös and Galácz 1998)
Haas et al. 1995; Balog et al. 1997; Haas and Budai 1999; Haas 2004; Péró and Haas 2004). While extended, thick shallow-marine sediments were characteristic of the Triassic, the Jurassic sediments had more localized appearances and featured thin slope and deep-marine sediments in the TDR. In the Late Triassic/Early Jurassic, the Dachstein platform was fragmented by the spreading of the Neotethys Ocean. During the intensive extensional tectonic movements, the pieces of the Dachstein platform were drowned and emplaced into a deeper environment. The blocks ended up at various depths: there were seamounts (topographic highs) and deeper-sea basins. On the top of the seamounts, condensed red nodular limestones and hard grounds were deposited. Hierlatz Limestone (brachiopod–crinoidal limestone) accumulated at the bottom of the slope around the heights. In the deeper basin, deep-sea sediments were formed: Ammonitico Rosso-type limestone, cherty limestones, and chert (Konda 1970; Galácz and Vörös 1972; Galácz et al. 1985; Galácz 1988; Haas 1988; Vörös and Galácz 1998; Haas 2012; Fig. 2). During the Toarcian Oceanic Anoxic Event (T-OAE; Jenkyns 1988; Jenkyns et al. 1991), black shale/gray shale-hosted (Polgári et al. 2016) manganese-rich sediments were deposited in one of the aforementioned drowning basins (Szabó-Drubina 1959; Cseh Németh 1967; Szabó et al. 1981; Kaeding et al. 1983; Jenkyns et al. 1991; Polgári et al. 1991, 2000, 2004, 2012; Weiszburg et al. 2004; Szabó 2006; Horváth et al. 2014; Zajzon et al. 2014).

Analytical methods

The investigations required for this study were carried out in the Institute of Mineralogy and Geology, University of Miskolc.

X-ray powder diffraction (XRPD) measurements were carried out using a Bruker D8 Advance diffractometer. The measurement parameters were the following: CuKα radiation, 40 kV accelerating voltage and 40 mA tube current, parallel beam geometry obtained with Göbel mirror in reflection geometry, Vantec-1 position-sensitive detector with 1° detector window opening, and 0.007° 2θ/14 s goniometer speed. The instrumental parameters of the oriented clay mineral measurements were the same; the only difference was a longer (29 s) goniometer speed. For the clay mineral investigation, the carbonate minerals within the footwall rock were dissolved in 5 wt% acetic acid (CH₃COOH), and the <2 μm grain size was collected by gravitational separation in a distilled water column. The settling time was determined using Stokes’ Law (Gibbs 1965). To prepare the oriented specimens, the pipette-on-glass-slide method (Brindley 1961; Kittrick 1961; Gibbs 1965) was used. From every sample, two specimens were prepared: both were measured in air-dried state, after which one was treated by ethylene glycol (vapor method by Brunton 1955) to determine the swelling value. The other sample was measured after 350 and 550 °C heat treatment to observe structural collapse.

The X-ray fluorescence (XRF) measurements were carried out using a Rigaku Supermini 200 wavelength-dispersive XRF spectrometer that was used with a 200 W Pd X-ray tube, 50 kV acceleration voltage, and 4 mA tube current. The calibrated
measurements were controlled with ZSX software (SamX, Lavardens, France). The analysis was run on pellets (32 mm diameter) made from 1 g of Cereox (C$_{38}$H$_{76}$N$_2$O$_2$) powder and 4 g homogenized bulk sample. The homogenized mixture was pressed with a 25-ton pneumatic press (Vaneox, Bedburg-Hau, Germany).

For the XRF and bulk XRPD, the samples were crushed: first 6 g bulk sample was crushed in a tungsten carbide breaking mortar, then the material was pulverized in a porcelain mortar, after which 1 g powder was separated and grinded to <5 μm in an agate mortar for XRPD measurements; from the remaining material, 4 g was weighted for XRF.

Fig. 2
The evolution of the Transdanubian Range (1, 2: after Haas 2004; 3, 4: after Péró and Haas 2004)
The scanning electron microscopy (SEM) measurements were completed on a JEOL JXA-8600 Superprobe EMPA (Tokyo, Japan). The parameters of the measurements were 20 kV acceleration voltage and 20 nA beam current.

**Samples**

The investigated samples were selected from seven boreholes: B-180, KT-96, U-159, U-163, U-178 U-363, and UB-43 (Fig. 3). Instead of whole cores, only a representative sample was preserved from each lithology in the local core storage facility. This resulted in variable amounts of samples and a variable sample frequency in the different boreholes: sometimes only small pieces of rocks were preserved within a few meters interval; in other places, samples were available from every 20 cm of the core. During the sample/core hole selection, the best resolution was the main criterion,
within the footwall and in the ore body as well (Bíró et al. 2009; Bíró 2014). From the 7 boreholes, 40 samples were investigated (Table 1). 0 m means the beginning (base) of the T-OAE. To identify the beginning of the T-OAE, we based ourselves on the study of Bíró (2013), who reevaluated all of the archive data: drilling descriptions for all core holes, with modern lithology and formations. Where the lithology changes, from limestone to clay/Mn-oxides/Mn-carbonate, there is also a formation change from the footwall formation to the Úrkút Manganese Ore Formation. We have placed

| No. of borehole | Relative depth (m) | Lithological description                      |
|-----------------|-------------------|---------------------------------------------|
| B-180           | 0.0 to 1.8        | Brown radiolarian clay                      |
|                 | −1.4 to 0.0       | Cherty clay                                 |
|                 | −19.6 to −1.4     | Limestone with crinoids and chert           |
| Kt-96           | 0.0 to 0.7        | Yellow radiolarian clay                     |
|                 | −11.6 to 0.0      | Light red limestone with tripoli lentil     |
|                 | −14.6 to −11.6    | Red limestone with crinoids                 |
| U-159           | 5.8 to 6.6        | Green manganese carbonate                   |
|                 | −0.8 to −6.3      | Red limestone with intraclasts              |
|                 | −6.3 to −13       | Red limestone with chert and tripoli        |
|                 | −13 to −19.6      | Red chert                                  |
| U-163           | 3.35 to 3.8       | Green manganese carbonate                   |
|                 | 1.6 to 2.1        | Green manganese carbonate                   |
|                 | 1.25 to 1.6       | Gray/green manganese carbonate              |
|                 | 0 to 1.25         | Light red limestone with intraclasts        |
|                 | −2.75 to 0.0      | Light red limestone with Mn-oxides          |
|                 | −2.75 to −5.55    | Red limestone                              |

Table 1
Description of the investigated samples
the beginning of the T-OAE at that point. Negative meters of depth signify that the sample originates below the T-OAE, whereas positive meters imply that the sample is younger than the beginning of T-OAE.

| No. of borehole | Relative depth (m) | Lithological description                                      |
|-----------------|--------------------|----------------------------------------------------------------|
| U-178           | 6.1 to 6.7         | Grayish green clayey manganese carbonate                      |
|                 | 3.8 to 5.6         | Black/gray clay shale                                         |
|                 | 3.1 to 3.3         | Green manganese carbonate                                     |
|                 | 1.6 to 2.1         | Green manganese carbonate                                     |
|                 | 0.6 to 1.1         | Brown manganese carbonate                                     |
|                 | 0.0 to 0.6         | Black/gray shale                                              |
|                 | –1.9               | Light red limestone with Mn-oxides veins                      |
|                 | –4.9               | Light red limestone with Mn-oxides veins                      |
|                 | –9.2               | Light red limestone                                           |
|                 | –13.2              | Red/light red limestone                                       |
|                 | –14.8              | Red limestone                                                 |
| U-363           | 5.1 to 6.8         | Green manganese carbonate                                     |
|                 | 2.6 to 5.1         | Green manganese carbonate                                     |
|                 | 0.0 to 2.6         | Clayey manganese oxide                                        |
|                 | –0.4 to 0.0        | Red clayey limestone                                          |
|                 | –4.4 to –0.4       | Light red limestone with Mn-oxide veins                        |
|                 | –9.4 to –4.4       | Light red limestone with tripoli lentil                       |
| UB-43           | 2.1 to 4.0         | Brown clay with Mn-oxide                                       |
|                 | 1.1 to 2.1         | Brown clay with Mn-oxide                                       |
|                 | 0.0 to 1.1         | Brown clay with Mn-oxide                                       |
|                 | –0.8 to 0.0        | Hierlatz-type limestone with manganese oxide                  |
Results

Bulk mineralogy

The footwall of the manganese deposit is made up of limestone. The average mineralogy of the Pliensbachian rocks is >95 wt% calcite, < 5 wt% silicates (quartz 1%–3% and clay minerals 2%–4%). It was necessary to use acetic acid to dissolve the Pliensbachian rocks to extract the clay mineral content. We found that after the leaching, the residual content was only 1–5 wt% for samples far from the Pliensbachian/Toarcian border (>5 m), but that within 5 m of that border, the amount of the residual material increased. This result is based on weighing of the residual material from the XRF and XRPD data, except in samples that contained “tripoli.” Tripoli is pure recrystallized quartz, of a crumbly nature, which appears randomly within the footwall. Tripoli can be found in mm-dm-sized lenses. It does not dissolve in acetic acid, so the amount of tripoli is added to the common residual material. Approaching the T-OAE, the residual content increased (>10 wt%).

Clay minerals

The previous studies assigned the smectite only to the Toarcian sediments (black shale/gray shale and manganese ore; Kaeding et al. 1983; Viczián 1995; Polgári et al. 2013). Leskó (2014) provided the first proof that the smectite also appears just below the ore deposit (20–30 cm). This study led to the first widespread clay mineral study in the footwall of the Úrkút manganese ore deposit. In this work, smectite was found in the footwall (Pliensbachian limestone) in every investigated borehole and in every studied Pliensbachian rock. In Fig. 4, we present diffractograms from each borehole, from the deepest point where smectite occurred. Every sample shows some smectite content. There is a peak at ~15Å in air-dried state, which is shifted to 17–17.5Å during glycolation. After heating, this peak shifted to ~10Å. 060 reflections show peaks at 1.505Å and 1.490Å (Fig. 5). The results indicate a dioctahedral illite and smectite. The 15Å peak is not specific to dioctahedral smectite, but it is explained by the exchange of interlayer cations to Ca and the water layer surrounding them, which was caused by the Ca cation exchange of the acetic acid leaching (Moore and Reynolds 1997). Besides the smectite, there was illite in every sample (peak at 10Å in the untreated samples), and some samples had kaolinite content with a peak at 7.2Å, which was preserved after glycolation and heating to 350 °C but disappeared after heating at 550 °C (collapse of the kaolinite structure). Every sample contains randomly interstratified illite/smectite (Fig. 4). The deepest point below the deposit where smectite was found was ~15 m in the U-178 borehole. This is also almost the deepest point in the basin that was accessed by borehole from where a sample is available, and the Úrkút Manganese Formation is here almost complete. The entry level of the smectite could not be identified because the drilling was stopped after reaching the footwall and only a few meters of Pliensbachian limestone/marl were collected. The chemical composition of the
**Fig. 4**
XRPD results: smectite content in the footwall: smectite peaks shifting from 15Å to 17–17.5Å by ethylene–glycol treating

**Fig. 5**
Representative XRPD results from borehole U-159: 060 reflections of unoriented clay minerals (grain size less than 2 μm)
material was measured with SEM-EDS on dehydrated materials (after 550 °C heating). The investigated samples were the same as those measured with XRPD: oriented clay mineral samples, which were made by the pipette-on-glass-slide method (Brindley 1961; Kittrick 1961; Gibbs 1965). Table 2 shows representative SEM-EDS results from borehole U-178. Si, Al, and Mg can occur within the illite and smectite as well. Ca appears within the smectite because of the cation exchange of the acetic acid leaching; Na and K belong to the illite. Fe can belong to the smectite and to the nanocrystalline hematite, which colorizes the limestone. S and P probably relate to the nano-sized Fe-oxide, which gives the red color to the limestone. The Ca content increases with the approach to the Pliensbachian/Toarcian border. This suggests the smectite content increases when approaching the T-OAE. No conclusion can be deducted from the changes in the Fe-content, because Fe can be found within smectite and the nanocrystalline Fe-oxide as well. The chemical formula calculation of smectite is not possible because of the small grain size: during the SEM-EDS/EPMA measurements, we analyzed at least 2–3 phases (illite, smectite, and nanocrystalline Fe-oxide) at the same time. The clay minerals cannot be determined exactly in every detail (cation content) because during the acetic acid leaching, cation exchange and oxidation of Fe occurred.

### Geochemistry

XRF measurements were carried out on samples that came from boreholes that produced at least five usable samples. These were the U-159, U-163, U-178, and

| U-178 | Na$_2$O | MgO | Al$_2$O$_3$ | SiO$_2$ | P$_2$O$_5$ | SO$_3$ | K$_2$O | CaO | TiO$_2$ | Fe$_2$O$_3$ |
|-------|--------|-----|------------|--------|-----------|-------|--------|-----|--------|------------|
| –7.8 m | 0.41   | 5.22| 22.09      | 58.31  | 1.49      | 0.21  | 4.20   | 2.27| 0.48   | 4.99       |
|       | 0.66   | 5.74| 23.64      | 58.59  | 1.56      | 0.53  | 3.27   | 2.06| 0.24   | 3.71       |
|       | 0.59   | 5.42| 23.21      | 58.85  | 1.57      | 0.38  | 3.84   | 2.25| 0.34   | 3.55       |
| –8.6 m | 0.72   | 5.75| 23.54      | 58.12  | 1.49      | 0.85  | 3.26   | 1.99| 0.40   | 4.21       |
|       | 0.66   | 5.75| 23.64      | 58.59  | 1.56      | 0.87  | 3.27   | 2.06| 0.24   | 3.72       |
|       | 0.38   | 5.39| 23.31      | 58.45  | 1.54      | 0.39  | 3.39   | 2.28| 0.38   | 4.49       |
| –13 m  | 1.62   | 4.89| 23.97      | 57.48  | 0.22      | 0.31  | 3.12   | 1.00| 0.70   | 6.68       |
|       | 1.09   | 4.40| 24.14      | 58.01  | 0.23      | 0.19  | 3.30   | 1.05| 0.71   | 6.89       |
|       | 0.87   | 4.47| 24.15      | 58.17  | 0.17      | 0.45  | 3.16   | 1.01| 0.70   | 6.85       |

The negative number means the distance below the beginning of the T-OAE.
U-363 boreholes. We focused on redox elements (Mn, Fe, Co, Ni, and Cr; Takeno 2005) because the deposit accumulated during an anoxic event: the environment was suboxic (oxygen-poor) or anoxic. These elements show high mobility under redox conditions. The main and trace element investigation shows that the studied elements were enriched only in the Toarcian sediment (during the T-OAE; Fig. 6). The U-363 borehole is an exception because there is a little enrichment just before the T-OAE. It was caused by thin Mn-oxide veins, which penetrated the footwall limestone. We found manganese veins only close to the Pliensbachian/Toarcian border (>~5 m). The Mn-oxide occurrences are mm-thick, irregular zigzag-shaped veins, which have infiltration zones. In the Pliensbachian footwall, the amount of Mn, Fe, and trace

Fig. 6
XRF results of investigated samples of four boreholes: geochemical differences between Pliensbachian and Toarcian layers
elements is negligible. However, from the beginning of the T-OAE, the sediments as well as the black/gray shale are enriched in these elements.

Tripoli

Tripoli is a “light-colored porous friable siliceous sedimentary rock/an incomplete silicified limestone from which the carbonate has been leached” (Jackson 1997). The Káváshegy Limestone Unit (KLU; Isztimér Limestone Formation) appears in many places just below the Toarcian strata (Bíró 2014). The KLU is enriched in red chert and has a limited range of occurrence: it appears only in the Bakony Mts. (Gyalog 1996). Tripoli was found in some footwall samples: it is visible or it was identified after acetic acid leaching. Its appearance is lenticular or massive. Tripoli is the recrystallization of the former SiO₂-bearing fossils (e.g., sponges and radiolarians; Polgári et al. 2005). The XRPD measurement and the SEM investigation showed that the measured material in our sample was pure quartz (Fig. 7).

SEM investigation was carried out on polished surfaces of bulk rock and acid-treated, polished-surface tripoli.

Fig. 7
SEM-BSE picture of the tripoli. Recrystallized SiO₂ material within the footwall limestone
Discussion

The Mn-oxide veins within the footwall could have been formed by the acetic-leaching process mentioned first in the study of Polgári et al. (2005) and based on all our observed data. The Mn-oxide veins appear right below the deposit, penetrating only few meters into the footwall. The acidic solution cannot penetrate deep into the footwall limestone because the carbonate rocks increase the pH; thus, the dissolving capability of the solution decreases. These observed veins and their above-proposed formation mechanism do not eliminate completely the possibility of tectonically formed Neptunian dikes elsewhere in the basin, which could well fit in the local, general tectonic environment observed, for instance in Eplény (Papp et al. 2015).

Smectite formation could take place in many ways: (a) low-temperature formation through the aging of artificially synthesized, freshly prepared hydroxide–silica precipitates (Harder 1972, 1976), whose process could happen in nature as well; (b) precipitation from hydrothermal solution (Bischoff 1972; Cole 1983; Lantenois et al. 2008); (c) alteration of volcanic rock and volcanic glass within a marine environment (Seyfried et al. 1978); and (d) precipitation from low-temperature fluids within basalt fractures (near the oceanic spreading zones; Hein and Scholl 1978; Bustillo and Martinez-Frias 2003).

In the entire TDR, there is no evidence for magmatism. There are no igneous rocks, tuffs, or volcanic glasses in the Pliensbachian or Toarcian sediments, so the third and fourth ways of smectite formation can be excluded.

Kaeding et al. (1983) refer to a hydrothermal ore and smectite formation. The main evidence was (a) the unique clay mineral content: smectite was present only within this deposit in the entire TDR in that period; (b) any trace element enrichment (e.g., Co).

We support low-temperature smectite formation through the aging of freshly formed hydroxide–silica precipitates, as described by Polgári et al. (2013); however, they identified the smectite only within the ore. The hydroxide–silica material could have been derived from the SiO2-containing fossils. This SiO2 material was recrystallized later and we can identify it now as tripli.

There are contradictions with previous studies that assumed that smectite appears only in the Toarcian sediments within the deposit (Kaeding et al. 1983; Viczián 1995; Polgári et al. 2013). In addition, the explanations deal with two smectite formation models: precipitation from hydrothermal solution and low-temperature formation through the aging of freshly formed hydroxide–silica precipitates. There are some trace element enrichments within the deposit, which support the hydrothermal formation mechanism (Kaeding et al. 1983).

We have carried out the first widespread clay mineral study that investigated the footwall (Pliensbachian) limestone/marl of the Úrkút manganese ore deposit. We found smectite in every studied core and footwall sample, so the results of Leskó (2014) could be confirmed: smectite appears everywhere below the deposit.

In our clay mineral study, we observed that by heating the sample at 550 °C, no residual peak appeared at 13Å, indicating the absence of chlorite. This proves the lack
of terrigenous sediments, which in turn confirms that the deposit was located far from the continent (Fig. 2; Meunier 2005).

Our geochemical study shows significant differences within the Pliensbachian and Toarcian sediments. In the Pliensbachian limestone/marl, there is no Mn, Fe, or trace element enrichment, but the Mn, Fe, and redox-sensitive elements show increasing values in the Toarcian rocks.

From these new results, it could be concluded that a hydrothermal origin should be excluded. If the trace element enrichment and the smectite content are connected (Kaeding et al. 1983), smectite should not appear within the footwall, as there were no elevated trace elements detected. There are two possibilities that could explain the relationship between the smectite appearance and the ore-forming process, which caused Mn, Fe, and trace element accumulation. (a) They are independent of each other; different processes caused the presence of smectite within the footwall and in the deposit and led to the ore accumulation. (b) The smectite (within the Pliensbachian and Toarcian strata) and the Toarcian Fe–Mn and trace element enrichment are related to each other, and both formed by a single process, which could have had several phases: the first caused smectite formation during the Pliensbachian; and later during the Toarcian, the smectite formation was continued and the ore accumulated as well.

We support the second possibility: several details point to a complex flow system (upwelling controlled the ore formation): asymmetric location of the manganese-rich T-OAE-related sediments (Vörös and Galácz 1998); local appearance of the SiO₂-rich footwall limestone, where the SiO₂ could have come from radiolarians (Polgári et al. 2005); locally, high amount of radiolarians could represent an upwelling system (Smith 1968; Hartline 1981; Molina-Cruz 1984) and common details with other upwelling-controlled sedimentation within the TDR: radiolarite (tripoli)-containing carbonate, isolated areas connected to the opening of the Neotethys Ocean, and asymmetric sedimentation (Budai et al. 2017).

Conclusions

Our new data support the lack of any igneous event during Pliensbachian and Toarcian times in the Úrkút Basin, which was mentioned in previous studies.

The smectite content is not only limited to the ore beds but appears within the footwall in the Úrkút Basin as well. This means that the process causing the smectite precipitation began earlier, not in the Toarcian but in the Pliensbachian. If we assume that the smectite formation belongs to the ore-forming process, as the previous studies mentioned as well, we must explain why there is not any manganese accumulation during the Pliensbachian, but only in the Toarcian. The explanation is to be found in the environment changes between the Pliensbachian and Toarcian (Jenkyns 1988; Jenkyns et al. 1991). The sharp geochemical differences between the footwall rocks and the ore deposit were caused by the changing environment, particularly in oxygen fugacity.
Summary

This paper describes the first study of clay mineral content of the footwall of the Úrkút manganese ore deposit. The investigation covered the entire area of the deposit: the measured samples also included old mining areas where production ceased more than 30–40 years ago.

The unique smectite content of the Úrkút manganese deposit’s footwall rocks and the geochemical differences between the Pliensbachian and Toarcian sediments were presented in this paper. Previous studies assumed that in the entire TDR, from the beginning of the Triassic until the end of the Toarcian (Jurassic), smectite appears only in the T-OAE-related Úrkút manganese deposit. However, all the investigated Pliensbachian footwall samples contain smectite. We identified dioctahedral smectite, illite, and in some samples minor kaolinite.

We explained the contradictions between the previous results (based on former observations, such as smectite occurrence being confined only to the ore deposit, and the enrichment of hydrothermal trace elements such as cobalt) and our new data. We assume that the ore accumulation was controlled by a complex flow system (upwelling-controlled ore formation).

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