An Evaluation of the Shielding Effectiveness of Tellurite Glass with Composition 85TeO$_2$-5Nb$_2$O$_5$-5ZnO-5Ag$_2$O for Diagnostic Radiology Application

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Author’s contribution

The sole author designed, analysed, interpreted and prepared the manuscript.

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ABSTRACT

The metal oxide glasses have attracted huge interest as promising types of shielding materials to replace the toxic, heavy and costly conventional shielding materials. In this work, the physical and the shielding effectiveness of Tellurite glass sample (S1) contain host metal oxides (85TeO$_2$-5Nb$_2$O$_5$-5ZnO-5Ag$_2$O) were evaluated at photon energies range between 15keV and 1MeV. The shielding parameters of the proposed glass system such as linear attenuation coefficients, HVL, MFP, $Z_{\text{eff}}$, and $N_{\text{eff}}$ were evaluated. The proposed samples showed a superior performance at the diagnostic energy range between 40 and 90 keV and a comparable shielding effectiveness above 90keV when compared with other commercial standard shielding materials.

Keywords: Tellurite glasses; mass attenuation coefficient; linear attenuation coefficient; half-value layer; mean free path; radiation shielding.

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1. INTRODUCTION

Since the discovery of x-ray by Roentgen in 1895 and the radioactivity by Henri Becquerel in 1896, the radiation applications such as nuclear power, medical imaging, cancer treatment and nuclear engineering are dramatically increased over the past decades [1-3]. The improper use of radiation can lead to serious injuries. The severity of radiation injuries depends on the radiation exposure dose rate. The radiation risk and injuries in different application were widely addressed by several researcher [4-8].

The use of ionizing radiation required safety standards to be establish and implement to ensure the protection of people and environment, many standards for radiation protection and safety were established [9-12]. Exposure time, distance from the source and shielding are the basic principles of protection considered when dealing with ionizing radiation. Shielding is most important consideration for any facility that perform diagnostic or therapy procedures. Many factors affecting the effectiveness of a shielding materials used to protect people working in radiation facilities such radiation energy, radiation type, the thickness of shielding material. The denser the material the more effective shielding materials against X-rays and gamma rays. Lead is the most common material use in most of radiation applications as shielding material due to its high atomic number. Although, lead is the most effective material in attenuating gamma and X-ray photons but can cause pollution through release of lead particles and can affected all the system body [13-14].

Due to the toxicity, high cost, and heavy weight of lead and lead glass materials several studies were conducted to develop alternative materials that replace lead such as tellurite-base glasses, phosphate-based materials, metal alloy, polymers [15-28]. These studies showed a good shielding performance in terms of high shielding efficiency without loss of transparency, good physical properties, thermal stability and good optical properties.

The shielding effectiveness have been studies in terms of the physical properties, radiation shielding parameters, LAC, MAC, HVL, MFP, Zeff and Neff. Al-Hadeethi et.al [24] have studied the x-ray photons attenuation characteristics for two glass-based systems (B2O3 -TeO2 – TiO3 and PbO-ZnO-TeO2-B2O3) at photon energies ranging from 30 to 80kVp. Their results showed that the increase of Tellurium dioxide (TeO2) concentration increases the attenuation coefficients of the glasses system with a decrease in the half-value layer especially at photon energy range between 70 and 80keV. Mhared et.al [25] have studied the shielding effectiveness of lithium-magnesium-borate glasses with Thulium oxide (Tm2O3). Their results showed that the attenuation coefficients increase with increasing Tm2O3 concentration. Lakshminarayana et.al [26] have studied the radiation shielding effectiveness of borosilicate glasses doped Tm3+ ions for gamma application. Their results indicates that the sample contain highest mole concentration of Tm2O3 has the greater ability to attenuate gamma-rays.

In this work, the shielding effectiveness of Tellurite based-glass sample (S1) was investigated at photon energy range between 15keV and1MeV. The radiation shielding parameters of the prepared glass system such as LAC, MAC, HVL, Zeff, Zeq, MFP, Neff and EBF were calculated using the online developed software (Phy-X/PSD) [27]. The results of shielding parameters were compared with other commercial shielding materials commonly used in photon applications.

2. THEORY AND METHOD

Tellurite glasses samples contain different oxides(S1) (85TeO2-5SnO2-5ZnO-5Ag2O) were prepared by putting the raw material in Platinum crucible in the heating furnace at a temperature in range from 850 to 950 °C for 30 min. The melting material was stirred to increase the viscosity before cast in the brass mold. The prepared sample was put in the annealing furnace for 2h at 320 °C. The sample density was measured using Archimedes’ Principle. Table1 shows the density, molar weight (MW) and the chemical compositions of the prepared sample.

| Table 1. The Physical parameters and chemical compositions of the proposed glass (S1) |
|---------------------------------|----------|----------|----------|-----------|----------|----------|----------|
| Sample code | Mw(g/mol) | Density(g/cm3) | ± 0.04 | TeO2 | Nb2O5 | ZnO | Ag2O |
|----------|----------|----------------|--------|------|------|-----|------|
| S1       | 164.61   | 5.3744         | -      | 85   | 5    | 5   | 5    |
The average molar weight of mixtures $\bar{M}$ can be calculated using mole by fractions $x_i$ and molar masses $M_i$ of the constituent elements of the component and their [28]:

$$\bar{M} = \sum x_i M_i,$$

where $x_i$ is the molar fraction of each component $i$, $M_i$ is the molecular weight of the sample.

The molar volume ($V_M$) of glass material can be calculated using the following equation [28]:

$$V_M = \frac{\bar{M}}{\rho},$$

where $M$ is the average molar weight of the sample, $\rho$ is the density of the sample.

$$\text{OPD} = 1000 \sum x_i n_i \left(\frac{1}{V_M}\right),$$

where $V_M$ is the molar volume of the glass materials, $x_i$ is a molar fraction, $n_i$ is the number of oxygen atoms in each oxide. The molar refraction of $R_m$ can be calculated using the following equation [28]:

$$R_m = \frac{n^2 - 1}{n^2 + 2} \times V_m,$$

The reflection loss, $R_L$, in percentage, can be calculated using the following equation [28]:

$$R_L = \left(\frac{n - 1}{n + 1}\right)^2,$$

where $n$ is the refractive index of the glass materials. The molar electronic polarizability ($\alpha_m$) can be calculated using the following equation [28]:

$$\alpha_m = \frac{R_m}{2.5},$$

The ionic concentrations of the glass samples are determined using the following relation [29]:

$$N = \frac{\text{mol} \% \text{ of Te} \times \text{glass density} \times \text{Avogadro's number}}{\text{Average molecular weight of glass}}$$

The polaron radius was calculated using the formula [29]:

$$r_p = \left(\frac{1}{2} \pi \frac{n}{6N}\right)^{1/3},$$

Where $N$ is the ionic concentrations.

Inter-ionic distance of the glass samples is given as [29]:

$$r_i = \left(\frac{1}{N}\right)^{1/3}$$

Where $r_i$ is the ionic concentrations.

The effectiveness of a shielding material can be investigated by the physical properties and radiation shielding parameters. The MAN, LAC, Zeff, Neff, HVL and MFP are the most important radiation shielding parameters that characterizing the effectiveness of the shielding materials. The cross section for scattering and absorption can be express in term of the total mass attenuation coefficient ($\mu/\rho$), which can be calculated using the program Xcom [30]. The mass attenuation coefficient of a compound can be computed using the following relation [30-34]:

$$\frac{\mu}{\rho} = \sum w_i \left(\frac{\mu}{\rho}\right)_i.$$
element, \( \Lambda_i \) is atomic weight of the ith atomic element.

The half-value layer (HVL) and the mean free path (MFP) are considered as important parameters for the estimation of the required effective shielding thickness for each photon energy. The HVL is the required thickness to reduce radiation intensity of the mono-energetic beam to its half value, while the MFP is representing the average distance between two successive interaction. These parameters can be computed according to the following relations [33-34):

\[
\text{HVL} = \frac{0.693}{\mu} \quad (15)
\]

\[
\text{MFP} = \frac{1}{\mu} \quad (16)
\]

Where \( \mu \) is the linear attenuation coefficient.

3. RESULTS AND DISCUSSION

Table 2 and 3 illustrate the calculated ion concentration, polaron radius, inter-ionic distance, oxygen packing density (OPD), the molar refraction (Rm) and the molar electronic polarizability (\( \alpha_m \)) of the prepared sample. As shown in Table 2, the smaller values recorded for the polaron radius and inter-ionic distance indicate an increase in polarizability, which leads to an increase in electrical conductivity of prepared glass system [29,35]. Table 3 shows comparison between the prepared glass sample and some commercially available glass shielding materials such as that developed by Schott Co., Germany standard shielding glass materials (RS 253-G18, RS 520 and RS 360) [36]. As shown in Table 3, the prepared sample recorded the highest values compared to the other glass systems. This indicates that the proposed glass has more non-bridging oxygen, which improve the stability of glass as host matrix for rare earth elements.

Fig. 1 shows the XRD pattern of prepared glasses. As shown in Fig. 1, the XRD pattern abroad defused scattering in 2θ range between 20 and 30, which indicate the amorphous glass nature of the prepared sample.

| Table 2. The Tellurite ion concentration, polaron radius, and inter ionic distance |
| --- |
| The concentration (N \( \times 10^{22} \)) ions/cm\(^3 \) (±0.01) | 1.67 |
| Polaron radius \( r_p \times 10^{-8} \), Å (±0.01) | 1.576 |
| Inter ionic distance \( r_i \times 10^{-7} \), Å (±0.01) | 3.911 |

| Table 3. The OPD, Rm and \( \alpha_m \) of the prepared sample prepared sample compared with RS 253-G18, RS 520 and RS 360 standard materials |
| Sample code | Refractive Index | OPD (mol/l) | Rm (m3) | \( \alpha_m \) (x10-24.cm3) |
| --- | --- | --- | --- | --- |
| S1 | 2.10 | 66.93 | 16.62 | 6.60 |
| RS253 G18 | 1.52 | 71.18 | 8.20 | 3.25 |
| RS360 | 1.62 | 38.22 | 13.32 | 5.29 |
| RS520 | 1.81 | 37.45 | 14.62 | 5.80 |

Fig. 1. XRD pattern of prepared glass
Fig. 3A and 3B show the computed half-value layers (HVL) of the prepared sample (85TeO2-5Nb2O5-5ZnO-5Ag2O) and some of the common commercial standard materials; RS 253-G18, RS 360, RS 520, Chromite, Chromite, Ferrite, Magnetite and Barite at energy range between 15keV and 1MeV. The LAC for the prepared glass sample were computed using the online developed software (Phy-X/PSD) [27]. The prepared sample recorded the highest values in the energies ranging from 40keV to 90keV as shown in Fig. 2A and 2B. For example, the value of linear attenuation coefficient recorded for the prepared sample was 13.29 cm⁻¹ compared to 1.148, 6.19, 10.63, 2.11, 2.45, 2.47 and 10.86 cm⁻¹ at 80keV, with percentage differences of 91.36%, 53.46%, 20%, 84.13% and 81.60%, 81.40% and 18.29 for RS 253-G18, RS 360, RS 520, Chromite, Chromite, Ferrite, Magnetite and Barite respectively. Above 90keV the prepared glass material recorded higher values than RS 253-G18, Chromite, Chromite, Ferrite, Magnetite and Barite, while the other standard glasses materials RS-520, RS-360 recorded slightly higher values than the prepared glass sample. The superiority of the prepared sample over all standard materials in the diagnostic energy range (40 to 90keV) is due to the fact that Tellurite glass doped with suitable modifier provide more bridging oxygen as host glass network, in addition to thermal stability, durability and good optical properties. These results are consistent with other findings [19, 20, 21].

Fig. 3A and 3B show the computed half-value layers (HVL) of the prepared sample (S1) compared with some commercially available shielding materials such as Schott Co. Germany standard shielding glass materials (RS 253-G18, RS 360, RS520) [36] and some of the common oxide used with concrete materials such as Chromite, Ferrite, Magnetite and Barite [37], at energy range between 15keV and 1MeV. As shown in the Fig. 3A and 3B the recorded HVL values of the sample is lower than the commercially available shielding materials in the energy range between 15keV and 90keV, which is expected due to the higher linear attenuation recorded for prepared sample compared with standard materials. For example, the computed MFP value of S1 was 0.052cm compared to 0.604, 0.056, 0.112, 0.065, 0.329, 0.281 and 0.064cm at 80keV, with percentage differences of 168.3%, 7.4%, 73.1%, 22.2%, 145.4% and 20.7% for the RS 253-G18, RS 360, RS 520, Chromite, Chromite, Ferrite, Magnetite and Barite respectively. The low recorded values are due to the high molecular weight and density of the prepared glass compared with the other standard materials. Above 90keV the prepared glass material shows superior effectiveness over RS 253-G18, Chromite, Chromite, Ferrite, Magnetite and Barite, while the other standard glasses materials RS-520, RS-360 recorded slightly lower values than the prepared glass sample.

As shown in Fig. 4A and 4B the prepared sample also recorded the lowest values of MFP compare with the RS 253-G18, Chromite, Chromite, Ferrite, Magnetite and Barite, and slightly higher than RS-520 and RS-360.
Fig. 3. The HVL for proposed glass sample (S1) compared with some common commercial shielding materials; at photon energy range between 15Kev and 1MeV (A); 80keV (B)

Fig. 4. The MFP for proposed glass sample (S1) compared with some common commercial shielding materials; at photon energy range between 15Kev and 1MeV (A); 80keV (B)

Fig. 5A and 5B shows the values of the total atom cross-section ($\sigma_a$) and total electronic cross-section ($\sigma_e$) as a function of photon energy for the prepared sample material compared with the commercially available shielding materials. The prepared sample recorded the higher values of $\sigma_a$ and $\sigma_e$ at diagnostic energy range.

Fig. 6A and 6B illustrated the effective atomic number ($Z_{\text{eff}}$) and effective electron numbers ($N_{\text{eff}}$) against photon energy (MeV) of the prepared glass sample. The radiation shielding values recorded for the prepared sample are comparable to the values provided for commercially available shielding materials. The maximum $Z_{\text{eff}}$ value of 49 was recorded at energy 40keV, while the minimum value of 23.1 is recorded at energy 1MeV, which indicates the better efficiency of the sample as a shielding material compared with the other samples. These results consistent with findings discussed before for linear attenuations.
Fig. 5. The ACS (A) and ECS (B) of proposed sample (S1) compared with standard materials at photon energy range between 15keV and 1MeV

Fig. 6. The Zeff(A) and the Neff(B) of proposed sample (S1) compared with standard materials at photon energy range between 15keV and 1MeV

4. CONCLUSION

Shielding and physical properties of a tellurite glass sample contain host metal oxides (85TeO₂ - 5Nb₂O₅ - 5ZnO - 5Ag₂O) were evaluated at photon energies range between 15keV and 1MeV. The shielding parameters of the proposed glass system such as linear attenuation coefficients, HVL, MFP, Zeff, and Neff were evaluated. The proposed samples showed a superior performance at the diagnostic energy range between 40 and 90 keV and a comparable shielding effectiveness above 90keV when compared with other commercial standard shielding materials. Each of the metal oxide selected for the preparation of the proposed glass material has its unique properties in term of the good physical, optical and shielding properties such as glass formation, thermal, radiation shielding effectiveness and
transparency, which make the prepared sample a promising glass material not only for shielding purposes but also for other medical applications.

COMPETING INTERESTS

Author has declared that no competing interests exist.

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