A comparative study on quality parameters of pumpkin, melon and sunflower oils during thermal treatment

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Abstract – Current paper reveals the impact of thermal treatment on the quality of two seed oils – pumpkin and melon compared to the quality of the most used oil – sunflower oil. Conventional and microwave heating were used for processing the oils. The duration of the thermal treatment was 9, 12 and 18 min for the conventional heating. The microwave heating was performed with two microwave powers of the equipment (600 W and 900 W) for 3, 6, 9 and 12 min. At every stage of the thermal processing were determined acid and peroxide value, the absorbance of the oils at 232 and 268 nm, tocopherol and fatty acid composition. It was observed that the degree of oxidation of the examined oils during microwave and conventional heating increased with the duration of the thermal process and the power of the microwaves. Also, the two methods of heating had a little impact on the processes leading to the formation of free fatty acids. Total tocopherols of the melon seed oil were more stable to thermal treatment. The amount of linoleic acid decreased in the pumpkin and sunflower oils during microwave treatment, while that of oleic and palmitic acid relatively increased. The biggest change in the fatty acid composition of both oils was found during microwave heating at 900W. The changes in fatty acid composition of thermally treated melon seed oil were insignificant. Overall, melon seed oil was observed to be more thermally stable than pumpkin and sunflower oils.

Keywords: pumpkin seed oil / melon seed oil / sunflower oil / thermal treatment / quality of oils

1 Introduction

Conventional heating is the most commonly used method in the world for food preparation. The thermal treatment is carried out at high temperature and in the presence of moisture and atmospheric oxygen, whereby the processes of oxidation and hydrolysis of the oils are activated. These reactions lead to a change in the functional, sensory and nutritional qualities of the processed foods and have great impact on human health. In the heat treatment of food products, microwave heating is also widely used (Lukešova et al., 2009). It is more efficient method and reduces the cooking time of the products. The application of microwave processing is increasing and therefore the attention of scientists is attracted to its impact on food products and their ingredients. However, the effect of microwave heating on the nutritional properties of the lipid fraction is insufficiently studied. During the heating of the oils are formed first primary products of oxidation (peroxides and hydroperoxides) and after that – secondary products such as aldehydes, ketones, etc. The deterioration of the vegetable oils and animal fats also depends on the content of the polyunsaturated fatty acids and the amount of free fatty acids increases during the microwave heating (Yoshida et al., 1990; Yoshida et al., 1992). According to Yoshida and Kajimoto (1989), total tocopherol content gradually decreases during the microwave treatment.

Although, some previous studies reveal the impact of thermal treatment of some common used plant oils (such as sunflower, soybean, peanut, rapeseed, corn, canola, etc.) (Lukešova et al., 2009; Vieira and Regitano-D’Arce, 1998; Hassanein et al., 2003), there is no information about the changes that occurs in melon seed oil subjected to conventional and microwave heating and the data about the pumpkin seed oil is rather scarce because both oils have limited application for cooking. Pumpkin and melon seeds have considerably high oil content over 50% for pumpkin seeds and 41–45% for melon seeds (Jiao et al., 2014; Petkova and Antova, 2015), which is a reason for their use as a rich oil source. The oils are abundant in essential fatty acids (such as linoleic and oleic acids) and tocopherols (mainly γ-tocopherol) (Jiao et al., 2014; Rabrenović et al., 2014; Petkova and Antova, 2015; Murkovic et al., 2004; Aktas et al., 2018). Though pumpkin and melon seed oils are rich of unsaturated fatty acids, they are reported to
have relatively high oxidative stability (the induction time of pumpkin and melon seed oils is 12.8–25.7 h and 4.28–5.9 h, respectively) (Szterk et al., 2010; Vidrih et al., 2010; Azhari et al., 2010; Mariod and Matthäus, 2008).

Jiao et al. (2014) performed a green microwave-assisted aqueous enzymatic extraction method for obtaining pumpkin seed oil and reported that this oil exhibited better oxidation stability compared to the Soxhlet-extracted oil. Szterk et al. (2010) also reveal that pumpkin seed oil is more oxidatively stable than rapeseed, linseed, primrose, camellina and borago oils, due to the presence of considerable amount of oleic acid and low amount of linolenic acid. This observation leads to the hypothesis that pumpkin and melon seed oils can be more stable when they are subjected to heating than some other oils used for cooking. Therefore, the aim of the present study was to examine the effect of microwave and conventional processing on pumpkin and melon seed oils. For that reason, have to be monitored the changes in the main physicochemical parameters (acid and peroxide value, the absorbance of the oils at 232 and 268 nm) as well as in the tocopherol and fatty acid composition of the oils, which were treated by conventional and microwave heating. The obtained results were compared to these of the most commonly used for cooking in Bulgaria sunflower oil in order to be concluded whether pumpkin and melon seed oils were suitable for thermal treatment.

2 Materials and methods

2.1 Materials

Pumpkin, melon and sunflower seeds were purchased by a specialized seed trading shop (Plovdiv, Bulgaria). The moisture content of the seeds was determined according to AOAC (2016) and was found to be as follows: 5.1% for pumpkin seeds, 6.0% for melon seeds and 6.5% for sunflower seeds. The oils were extracted from the seeds using Soxhlet extraction with hexane for 8 h (ISO 659:2014) and the yield was 51.5% for pumpkin seeds, 41.6% for melon seeds and 49.1% for sunflower seeds. The obtained oils were subjected to heating (conventional and microwave) with different duration. The microwave heating was performed into two powers 600W and 900W for 3, 6, 9 and 12 min in a microwave oven (LG Model MS1907C). The conventional heating was performed on electric cooktop with 1 burner for 9, 12 and 18 min.

All reagents used were purchased by Merck (KGaA, Darmstadt, Germany) and were with analytical grade.

2.2 Physicochemical characteristics

The physicochemical properties (acid and peroxide value) of the oils were analysed following the standard procedures by ISO (ISO 660:2009; ISO 3960:2017). The absorbance at 232 and 268 nm was determined using a Boeco S26 spectrophotometer after diluting the oil in isoctane (1/100, v/v) (ISO 3656:2011).

2.3 Tocopherol composition

Tocopherols and tocotrienols were determined directly in the oil by high performance liquid chromatography on a Merck-Hitachi (Merck, Darmstadt, Germany) instrument equipped with 250 × 4 mm Nucleosil Si 50-5 column and fluorescent detector Merck-Hitachi F1000. The operating conditions were mobile phase of hexane: dioxane: 96:4 (v/v) and flow rate 1 mL/min, excitation 295 nm, emission 330 nm. 20 μL 1% solution of crude oil in hexane were injected. Tocopherols and tocotrienols were identified by comparing the retention times with those of authentic individual ones. Reference individual pure tocopherols (DL-α-, DL-β-, DL-γ- and DL-δ-tocopherols with purity ≥98%) were purchased from Merck (Darmstadt, Germany) and Tocomin SupraBio was purchased from Carlson Laboratories, Inc. (Arlington Heights, IL, USA). The tocopherol content was calculated on the base of tocopherol peak areas in the sample vs. tocopherol peak area of standard tocopherol solution (ISO 9936:2016).

2.4 Fatty acid composition

Fatty acid composition of the examined glyceride oils was determined by gas chromatography (GC) (ISO 12966-1:2014). Fatty acid methyl esters (FAMEs) were prepared by transesterification of the oils with sulfuric acid in methanol (ISO 12966-2:2017). Determination of FAMEs was performed on HP 5890 gas chromatograph equipped with a 75 × 0.18 × 25 μm (film thickness) capillary Supelco column (SP-TM – 2560, Fused silica Capillary Column) and a flame ionization detector. The column temperature was programmed from 140 °C (hold 5 min), at 4 °C/min to 240 °C (hold 3 min); the injector and detector temperatures were set at 250 °C. Identification was performed by comparison of the retention times with those of a standard mixture of FAME (Supelco, USA 37 comp. FAME mix) subjected to GC under identical experimental conditions.

2.5 Statistical analysis

All measurements were performed in triplicate (n = 3) and the results were presented as mean value with the corresponding standard deviation (SD). Significant differences were determined by an analysis of variance (Duncan test) with a significance level p < 0.05 using IBM SPSS Statistics 19.

3 Results and discussion

3.1 Physicochemical characteristics

The measured temperatures of pumpkin, melon and sunflower oils after microwave and conventional heating are given in Table 1. The temperature increases with increasing the duration of heating and the power of the microwaves (p < 0.05). Higher temperature values for conventional heating compared to the microwave once were observed even for the same duration of processing.

Reactive free radicals are formed during the heating which rapidly interact with atmospheric oxygen to form hydroperoxides, and they can be determined with the peroxide value that depicts the degree of oil oxidation. The results about changes of the peroxide value of the pumpkin, melon and sunflower oils subjected to microwave and conventional heating are shown in Figure 1.
Changes in the peroxide value of pumpkin (A), melon (B) and sunflower (C) oils after conventional and microwave (MW) heating. Values are means ± SD (n = 3). Different letters in the same oil mean significant differences (p < 0.05) using Duncan test.

The peroxide value of the pumpkin seed oil decreased from 3 min to 9 min and again decreased to 8.8 and 8.6 meq active oxygen/kg at 12 min of heating (Fig. 1A). Their lower values in the studied periods can be explained by the fact that the primary products of the oxidation are hydroperoxides, which are unstable and once formed, they easily can be converted into secondary oxidation products. Similar results were obtained by Vieira and Regitano-D’Arce (1998) who monitored the changes in the peroxide value of microwave heated canola, corn and soybean oils. The peroxide value of the oils was found to have a maximum at 4 and 6 min, and then it decreased. In the canola oil, the peroxide value decreased at 6 min of heating, then increased at 20 min and after that was observed stabilization at 36 min. The peroxide value of the pumpkin seed oil increased significantly after conventional heating (from 11.8 to 22.8 meq active oxygen/kg; p < 0.05).

The amount of hydroperoxides during microwave heating of melon and sunflower oils increased with an increase of the duration of heating and the power of the microwaves (Fig. 1B and C). The maximum power (900W) produced a larger amount of hydroperoxides compared to microwave heating at 600W. The peroxide value of the conventionally heated melon and sunflower oils increased considerably (p < 0.05), but the latter possessed a higher degree of oxidation (from 17.0 to 33.7 meq active oxygen/kg). Melon seed oil had the lowest degree of oxidation after both microwave and conventional heating and its peroxide value met the requirements for this value for refined and unrefined vegetable oils (up to 10 and 15 meq active oxygen/kg, respectively) (Codex-Stan 210:1999).

The peroxide values of the three examined oils after conventional heating are higher compared to the microwave one. Probably, the higher temperature and longer duration of the heating influence the oxidation process in the oils. The obtained results differed from these reported by Poiana (2012) according to whom the microwave heating accelerated the lipid oxidation at a higher degree than the conventional treatment.

The accumulation of hydroxides is also associated with the simultaneous formation of conjugated dienes and trienes structures with an absorption maximum in the range of 230–234 nm and 265–270 nm. The absorbance of the heated oils diluted in isooctane at 232 and 268 nm is shown in Table 2.

The rate of oxidation processes during conventional heating was higher (the increase in the absorption values
Table 2. Changes in the absorbance of pumpkin, melon and sunflower oils at 232 and 268 nm after microwave (MW) and conventional heating.

| τ, min | MW heating 0.6W | MW heating 0.9W | Conventional heating 0.6W | Conventional heating 0.9W |
|--------|------------------|------------------|---------------------------|---------------------------|
| 0      | 0.937 ± 0.017b   | 0.937 ± 0.017a   | 3.050 ± 0.050                      | 3.050 ± 0.050                      |
| 3      | 1.053 ± 0.010b   | 1.150 ± 0.015b   | 3.056 ± 0.015c                   | 3.190 ± 0.011c                   |
| 6      | 1.012 ± 0.009b   | 1.165 ± 0.020b   | 3.214 ± 0.015c                   | 3.214 ± 0.015c                   |
| 9      | 1.117 ± 0.015f   | 1.196 ± 0.006c   | 3.282 ± 0.018d                   | 3.324 ± 0.020e                   |
| 12     | 1.142 ± 0.005f   | 1.339 ± 0.010b   | 3.256 ± 0.020d                   | 3.244 ± 0.014f                   |
| 18     | –                | –                | –                           | 3.536 ± 0.012e                   |
| 268 nm | 0.406 ± 0.006a   | 0.406 ± 0.006a   | 1.604 ± 0.008e                   | 1.604 ± 0.008e                   |
| 3      | 0.403 ± 0.003a   | 0.388 ± 0.006a   | 1.584 ± 0.010c                   | 1.554 ± 0.004c                   |
| 6      | 0.407 ± 0.007a   | 0.391 ± 0.005a   | 1.608 ± 0.008b                   | 1.556 ± 0.009c                   |
| 9      | 0.410 ± 0.005b   | 0.404 ± 0.004b   | 1.628 ± 0.005c                   | 1.560 ± 0.007c                   |
| 12     | 0.417 ± 0.007a   | 0.417 ± 0.007a   | 1.644 ± 0.004c                   | 1.602 ± 0.002a                   |
| 18     | –                | –                | –                           | 1.574 ± 0.005e                   |

Values are means ± SD (n = 3). Different letters in the same oil mean significant differences (p < 0.05) using Duncan test.
* No heating was performed at this period of time.

was about 70% for the pumpkin seed oil, 16% for the melon seed oil and 50% for the sunflower oil. The absorbance at 232 nm of the treated oil samples during microwave and conventional heating increased with the duration of the treatment and the power of microwaves. The microwave heating of melon and sunflower oils showed a minimal increase in absorbance at 232 nm — about 6–7% compared to its initial value, while in pumpkin seed oil the growth was about 20–40% (p < 0.05), indicating that the formation of conjugated dienes was accelerated.

Similar results were obtained in previous studies in which the thermal stability of rapeseed, sunflower, soybean, corn and canola oils were examined (Lukesova et al., 2009; Poiana, 2012; Vieira and Regitano-D’Arce, 1998). They showed that the microwave heating initiated lipid oxidation and the absorption at 232 nm increased due to the formation of conjugated dienes. It was also found that up to 10–12 min of heating sunflower, rapeseed, soybean and canola oils the change in the value of the absorption compared to that of the control sample was negligible, and then increased at the end of the thermal treatment.

The absorbance of the microwave heated pumpkin seed oil at 268 nm at medium and high microwave power had values close to the control oil (oil that was not processed thermally) (Table 2). The absorption of the pumpkin seed oil subjected to conventional heating increased and reached 0.573 at 18 min (p < 0.05). The values of absorption at 268 nm of microwave heated melon seed oil remained almost the same as the values of the control sample, while in the conventional heated oil was even lower than those of the control (Table 2). The absorption of microwave heated sunflower oil increased insignificantly with the duration of the heating, and the values were close to that of the control oil (Table 2).

It was observed that the values of the absorption at 268 nm of conventionally heated sunflower and pumpkin oils were relatively high, while the changes of the values of microwave heated oils were negligible. These results showed a higher oxidative stability of melon oil at both thermal treatments, while pumpkin and sunflower oils were more stable when exposed to microwave heating.

The obtained results were in agreement with these reported by Vieira and Regitano-D’Arce (1998), who established that the absorption at 270 nm increased after microwave heating of canola, corn and soybean oils. The increase of the absorption were from 0.762 (0 min) to 1.163–1.445 (12 and 20 min) for canola oil, from 1.587 to 2.096 (12 and 20 min) for corn oil and from 3.324 to 3.784–3.875 (12 and 20 min) for soybean oil, which depicted the formation of conjugated trienes in the oils.

Acid value (an indicator for the amount of the free fatty acids) of the pumpkin, melon and sunflower oils, that are conventionally and microwave heated, are shown in Table 3.

The acid value of pumpkin seed oil slightly increased when the time of microwave and conventional heating was extended and the power of the microwaves was increased. The acid value of the conventionally heated pumpkin seed oil was slightly lower, which indicated that during the microwave heating were produced more free fatty acids.

The acid values of microwave and conventionally heated melon and sunflower oils had the same values as the control oil, hence the thermal treatment of these oils did not initiate the formation of free fatty acids.

The obtained results differed from these reported by Vieira and Regitano-D’Arce (1998), who observed that the content of the free fatty acids increased after microwave treatment of corn oil — from 0.077 to 0.127–0.129% (12 and 20 min); and for soybean oil — from 0.075 to 0.107–0.132% (12 and 20 min).
3.2 Tocopherol composition

The changes in the total tocopherol content of the pumpkin, melon and sunflower oils, subjected to microwave and conventional heating are shown in Figure 2A–C. Total content of tocopherols decreased with increasing the time of microwave and conventional heating. They decreased from 983 to 566 mg/kg during microwave heating of pumpkin seed oil at 600W, and from 983 to 546 mg/kg at 900W \( (p < 0.05) \). During the conventional heating of the same oil, the tocopherols decreased to about 700 mg/kg after 9–12 min treatment of the sample, and then their amount sharply decreased to 437 mg/kg at 18 min \( (p < 0.05) \).

When melon seed oil was heated, the changes of total tocopherol content were minimal, both in the microwave (from 884 to 827 mg/kg at 12 min) and the conventional heating (from 884 to 701 mg/kg at 18 min).

The changes of tocopherol content of sunflower oil subjected to microwave heating was minimal (from 677 to 624 mg/kg), whereas in the case of conventional heating the amount of tocopherols decreased sharply after 9 min of heating of the oil \( /C_0\) from 548 to 312 mg/kg \( (p < 0.05) \).

The degradation of the tocopherols was to a lesser extent in the microwave and conventionally heated melon seed oil and in the sunflower oil subjected to microwave heating. Total tocopherols were reduced with 50% during the conventional heating of pumpkin and sunflower oils at 18 min. Considerably better preservation of the tocopherols after the microwave treatment was established and it was most likely due to the shorter time of the microwave heating. Yoshida et al. (1991, 1992) also established that prolonging the heating time affected total tocopherol content which amount decreased. According to these authors, a higher content of saturated fatty acids in the oils produces a larger amount of free fatty acids that have a destructive effect on tocopherols.

The unsaturated fatty acids predominated in the studied oils and the amount of free fatty acids in them was minimal so their impact on the destruction of tocopherols was smaller. Changes in the individual tocopherol composition of pumpkin, melon and sunflower oils subjected to microwave and conventional heating are presented in Table 4.

### Table 3. Changes in the acid value (mg KOH/g) of pumpkin, melon and sunflower oils subjected to microwave (MW) and conventional heating.

| Power of MW heating, W | Control oils (0 min) | Microwave heating | Conventional heating |
|------------------------|----------------------|-------------------|----------------------|
|                        | 3 min | 6 min | 9 min | 12 min | 9 min | 12 min | 18 min |
| Pumpkin seed oil       |       |       |       |       |       |       |       |
| 600                    | 3.7 ± 0.2 \(^a\)   | 3.9 ± 0.2 \(^{a,b}\) | 4.0 ± 0.1 \(^{a,b}\) | 4.3 ± 0.3 \(^{b}\) | 4.4 ± 0.2 \(^{b}\) | 4.0 ± 0.3 \(^{b}\) | 4.1 ± 0.2 \(^{b}\) |
| 900                    | 3.7 ± 0.2 \(^a\)   | 4.0 ± 0.2 \(^{a,b}\) | 4.3 ± 0.3 \(^{a,b}\) | 4.4 ± 0.1 \(^{b}\) | 4.6 ± 0.2 \(^{b}\) | 4.0 ± 0.3 \(^{a,b}\) | 4.1 ± 0.3 \(^{a,b}\) |
| Melon seed oil         |       |       |       |       |       |       |       |
| 600                    | 1.5 ± 0.2 \(^c\)   | 1.6 ± 0.2 \(^c\)  | 1.4 ± 0.1 \(^{c}\)  | 1.4 ± 0.1 \(^{c}\) | 1.4 ± 0.2 \(^{c}\) | 1.4 ± 0.1 \(^{c}\) | 1.4 ± 0.2 \(^{c}\) |
| 900                    | 1.5 ± 0.2 \(^c\)   | 1.4 ± 0.2 \(^c\)  | 1.6 ± 0.1 \(^{c}\)  | 1.4 ± 0.1 \(^{c}\) | 1.6 ± 0.2 \(^{c}\) | 1.4 ± 0.1 \(^{c}\) | 1.4 ± 0.1 \(^{c}\) |
| Sunflower oil          |       |       |       |       |       |       |       |
| 600                    | 0.3 ± 0.1 \(^d\)   | 0.3 ± 0.0 \(^d\)  | 0.3 ± 0.1 \(^d\)  | 0.3 ± 0.1 \(^{d}\) | 0.3 ± 0.0 \(^{d}\) | 0.3 ± 0.0 \(^{d}\) | 0.3 ± 0.0 \(^{d}\) |
| 900                    | 0.3 ± 0.1 \(^d\)   | 0.3 ± 0.1 \(^d\)  | 0.3 ± 0.0 \(^d\)  | 0.3 ± 0.0 \(^{d}\) | 0.3 ± 0.0 \(^{d}\) | 0.3 ± 0.0 \(^{d}\) | 0.3 ± 0.0 \(^{d}\) |

Values are means ± SD \( (n = 3) \). Different letters in the same oil mean significant differences \( (p < 0.05) \) using Duncan test.

![Fig. 2. Changes in total tocopherol content of pumpkin (A), melon (B) and sunflower oils (C) subjected to heating: 1 – 600W, 2 – 900W and 3 – conventional heating. Values are means ± SD \( (n = 3) \). * Different letters in the same oil mean significant differences \( (p < 0.05) \) using Duncan test.](image-url)
The qualitative tocopherol composition of the tested oils was preserved during the thermal heating and minimal quantitative changes were observed. The amount of α-tocopherol in pumpkin seed oil insignificantly increased after 9 min of microwave heating, but after that increased up to 4.2% (12 min at 900W). A considerable decrease of α-tocopherol (from 5.6 to 2.8%) was observed after conventional heating of the same oil (p < 0.05), while in melon seed oil the reduction of α-tocopherol was minor. The quantity of α-tocopherol in sunflower oil insignificantly decreased after microwave heating up to 9 min and after that increased (up to 96%), while during the conventional heating its amount slightly decreased up to 91.1% at the expense of the content of β-tocopherol. It was found that during microwave heating tocopherols are much more stable in oils which contained more unsaturated fatty acids than saturated, therefore no significant changes in the tocopherol composition of the examined oils were observed.

### 3.3 Fatty acid composition

Fatty acids containing 8–20 carbon atoms (C₈-C₂₀) were identified in all examined oils. In the control pumpkin seed oil were identified 10 fatty acids, in melon seed oil – 15, and in sunflower oil – 8. The composition of the oils predominantly consisted of linoleic, oleic, palmitic and stearic acids. Their content was as follows: linoleic (40.0, 54.6 and 51.8%, respectively), oleic (35.1, 20.7 and 35.5%), and palmitic (18.4, 15.1 and 12.1%), while those of stearic was found to be lower (5.9, 7.4 and 3.5%). The other fatty acids were present in small amounts up to 0.4%. The obtained results were in agreement with these reported by previous authors (Kim et al., 2012; Yanty et al., 2008; Petkova and Antova, 2015; Rabrenović et al., 2014; Codex-Stan 210:1999).

Fatty acid composition of pumpkin, melon and sunflower oils subjected to microwave and conventional heating was close to that of the control oils and major changes were observed in the quantity of the major fatty acids (Fig. 3).

### Table 4. Changes in the tocopherol composition of pumpkin, melon and sunflower oils during microwave and conventional heating.

| Tocopherols, %  | Control oils (0 min) | 600W  | 900W  | Conventional heating |
|----------------|----------------------|-------|-------|----------------------|
|                | 3 min | 6 min | 9 min | 12 min | 3 min | 6 min | 9 min | 12 min | 9 min | 12 min | 18 min |
| α-Tocopherol   | 5.6 ± 0.3³ | 6.7 ± 0.5⁴ | 6.4 ± 0.2³ | 5.9 ± 0.4b | 5.4 ± 0.2b | 7.5 ± 0.5⁴ | 8.3 ± 0.2c | 7.0 ± 0.5d | 4.2 ± 0.4⁴ | 5.7 ± 0.3³ | 4.2 ± 0.2f | 2.8 ± 0.2³ |
| γ-Tocopherol   | 89.9 ± 1.0d | 88.5 ± 0.8³ | 90.8 ± 0.5³ | 90.8 ± 0.7d | 90.8 ± 0.8³ | 87.1 ± 1.1³ | 88.5 ± 0.5d | 89.5 ± 0.7d | 92.9 ± 0.5³ | 90.4 ± 0.9³ | 92.0 ± 0.5³ | 93.9 ± 0.6d |
| γ-Tocotrienol  | 2.4 ± 0.4³ | 2.3 ± 0.1i | 1.4 ± 0.2ab | 2.0 ± 0.2³ | 2.0 ± 0.3³ | 2.5 ± 0.1³ | 1.6 ± 0.1³ | 1.9 ± 0.3³ | 1.0 ± 0.2³ | 2.1 ± 0.1³ | 1.7 ± 0.3³ | 1.1 ± 0.1i |
| δ-Tocopherol   | 2.1 ± 0.2³ | 2.5 ± 0.2³ | 1.4 ± 0.1i | 1.3 ± 0.3³ | 1.8 ± 0.4³ | 2.9 ± 0.2³ | 1.6 ± 0.2³ | 1.6 ± 0.1³ | 1.9 ± 0.3³ | 1.8 ± 0.2³ | 2.1 ± 0.1³ | 2.1 ± 0.1³ |
| Tocotrienol    | 5.6 ± 0.3b | 6.7 ± 0.5a | 6.4 ± 0.2a | 5.9 ± 0.4b | 5.4 ± 0.2b | 7.5 ± 0.5³ | 8.3 ± 0.2c | 7.0 ± 0.5d | 4.2 ± 0.4³ | 5.7 ± 0.3³ | 4.2 ± 0.2f | 2.8 ± 0.2³ |

Values are means ± SD (n = 3). Different letters in the same line mean significant differences (p < 0.05) using Duncan test.
The greatest changes in the fatty acid composition were observed in pumpkin and sunflower oils, while in melon seed oil, there were no significant changes in the main fatty acids. The decrease of total tocopherols in melon seed oil during heating was minimal and probably related to the preservation of linoleic acid in the oil. On the other hand, the changes in peroxide value of thermally treated melon seed oil was lower than these depicted in Codex-Stan (1999) (10–15 meq active oxygen/kg) which lead to formation of less amount of hydroperoxides that decomposes longer unsaturated fatty acids (Yoshida et al., 1991). Therefore, the variations in the main fatty acids of thermally treated melon seed oil were negligible.

4 Conclusion

The present study was first to monitor the changes that occurred in quality parameters (acid, peroxide value, absorbance at 232, 268 nm, tocopherol and fatty acid composition) of pumpkin and melon seed oils during conventional and microwave treatment and were compared to those of sunflower oil.

The degree of oxidation of pumpkin, melon and sunflower oils during microwave and conventional heating increases with the duration of the thermal process and the power of the microwaves. Conventional heating causes acceleration of the lipid oxidation. The highest thermal stability is observed in melon seed oil for all types of heat treatment, while pumpkin and sunflower oils are more stable during microwave heating. The thermal treatment of the oils has a little impact on the processes leading to the formation of free fatty acids.

The microwave heating has a little impact on decreasing the content of tocopherols compared to conventional heating. Total tocopherols of melon seed oil seem to be more stable to thermal treatment (microwave and conventional). There are no considerable changes in the tocopherol composition of the examined oils during heating.

The decrease of the quantity of linoleic acid in pumpkin and sunflower oils during microwave heating was at the expense of the increase of the content of saturated fatty acids and this could be explained with the reactions described by Farag et al. (1992). The authors reported that oxidative degradation and production of short-chain acids were observed during microwave and conventional heating of oils. According to Yoshida et al. (1991), tocopherols are more stable during microwave heating in the longer fatty acids than they are in the shorter ones. The decrease of total tocopherols in melon seed oil during heating was minimal and probably related to the preservation of linoleic acid in the oil. On the other hand, the changes in peroxide value of thermally treated melon seed oil was lower than these depicted in Codex-Stan (1999) (10–15 meq active oxygen/kg) which lead to formation of less amount of hydroperoxides that decomposes longer unsaturated fatty acids (Yoshida et al., 1991). Therefore, the variations in the main fatty acids of thermally treated melon seed oil were negligible.

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Conflicts of interest. The authors declare that they have no conflicts of interest in relation to this article.

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