Influence of enzymes on the oil extraction processes in aqueous media

Guillaume RICOCHON
Lionel MUNIGLIA
LIBio,
Nancy Université INPL,
2, avenue de la Fôret-de-Haye,
BP 172, 54505 Vandœuvre-lès-Nancy,
France
<lionel.muniglia@ensaia.inpl-nancy.fr>

The idea to develop an aqueous extraction process (AEP) to produce oil from oilseeds was born in the 1950s. In those days, such a process seemed cheaper and less dangerous than the processes of extraction by solvent which supplied then the best yields. Unfortunately, 50 years later, the yields obtained by the AEP have not still reached the maximal efficiency that would be necessary for their economic viability. However, the constantly growing interest of manufacturers and authorities for the safety of employees, quality and environmental protection has managed to maintain AEP as a topic which can sometimes compete with the other usual processes—i.e., for specific oils with better added value. The AEP can be distinguished by its use of water as a single media of extraction. Moreover, the principle of extraction is radically different from the one based on the use of solvent. These principles were studied by Johnson and Lusas and resumed by Rosenthal (Johnson and Lusas, 1983; Rosenthal et al., 1996).

The processes of extraction by solvents are based on their capacities to dissolve oils and to extract them from the complete seed. In the AEP, oil does not have the same affinity with water so it cannot be dissolved in this solvent, thus leading to low yield. The extraction of oil in aqueous media is based on the opposite phenomenon to the one operated by extraction by solvent, i.e., the insolubility of oil in water. The objective is to dissolve in water the soluble compounds of the seed which retain the oil in a closed structure to naturally keep it in the seed. In theory, the extraction yields obtained in the aqueous phase are facilitated by all techniques (using enzymes, grinding, etc.), allowing the dissolution of the soluble compounds in the water to increase. It is important to know the constituents of the plant cell walls to be able to choose which enzymes are to be used, and then among the families of chosen enzymes, to find the most efficient ones to finally define which factors allow to obtain the best yields.

Cell wall constituents and specific enzymes

To extract the lipid reserves stored in cells, it is necessary to be able to cross several barriers: first the extra cellular walls (or secondary cell walls), then the cell wall, and finally oleosomes. Each cell wall has its own constituents, sometimes organized in a complex structure, and synthesized and degraded in a natural manner by specific enzymes. For example, the secondary cell walls of rapeseeds are constituted of 39% pectins, 29% hemicelluloses, 22% cellulose, 8% of arabinogalactans (Dominguez et al., 1994) and constitute 20% to 28% of the total seed. The primary cell wall contains 10% of glycoproteins which are very rich in hydroxyproline, at first called extensins but called today HRGP (HydroxyProline Rich Glycoprotein).

Cellulase is a linear chain of several hundred to over nine thousand of β-(1→4)-linked D-glucose units. In the cell wall, the molecules of cellulose are assembled in parallel rows: microfibrils with a diameter from 5 to 12 nm. They are constituted from 36 to 1,200 molecules of cellulose maintained together by hydrogen bonds between the hydroxyl groups of the nearby glucose residues. They are between 5 and 15 nm wide and separated from one to another by 20-40 nm. Crystalline and amorphous domains can be observed.

Enzymatic degradation: the complete degradation of cellulose requires three types of enzymes: two cellulases (EC 3.2.1.4 and EC 3.2.1.91) and a β-glucosidase (EC 3.2.1.21). The first cellulase, called 1,4-β-cellobiase, is able to hydrolyse the intermolecular β-(1→4)-glucosidic bonds. The second cellulase can hydrolyse the cellulose from the extremities of glucosidic chains; they consequently form either glucose or cellobiose. Finally, β-glucosidases can hydrolyse the cellobiose molecules, forming some glucose, to eliminate its inhibition on the other activities.

Hemicelluloses are linear or branched out polysaccharides bound to the cellulosic microfibrils by hydrogen bonds or connected to the lignin by covalent bonds, so they form a complex and solid structure around plant cells. This complex polymer requires, as for the degradation of the cellulose, a set of enzymes able to work together. For example, the hemicelluloses of soybean and black locust (Shallom and Shoham, 2003) published a review detailing the mode of action of the following hemicellulases:

- Xylanases (EC 3.2.1.8) hydrolyse β-1,4 linked chains of xyloses, producing small xylo-oligomers. Lateral chains of glucuronic acids or of arabinoses fixed to this xylosidic skeleton can hide the action of these enzymes. Nevertheless, Hubert et al. identified a xylanase which need the presence of 4-O-methyl-glucuronic acid to spot and break β-1,4-xylosidic links.

Abstract: The methods of oil aqueous extraction process (AEP) assisted by enzymes are, over the last 50 years, an alternative designed to replace traditional methods of extraction using organic solvents. To extract the oil using an AEP, the use of specific enzymes, able to hydrolyse some or all components of seeds, can significantly increase the yields of extraction. Hydrolyzing the different constituents of cell walls (cellulose, hemicellulose, pectins, proteins, etc.), enzymes are able to enhance the liberation of the oil. A number of physico-chemical parameters must also be considered for the better expression of the enzymatic mixture, while maintaining the quality of oils and meals. This article presents the various factors influencing the release of oil in aqueous media and the main results obtained by this process on various substrates.

Key words: oil aqueous extraction process, enzymes, quality
Proteins

Proteases correspond to any enzyme which conducts proteolysis, that is to say, any enzyme which begins protein catalysis by hydrolysis of the peptide bonds linking amino acids together in the polypeptide chain. There are 5 types of proteases, classified according to their mechanism of action and amino acid (serine, threonine, aspartate, cysteine) or a metallic atom directly implicated in the mechanism. In theory, using proteases could be interesting for extracting oil. It would allow the proteins of cell wall structures to break, as well as olesins, which stabilize oleosomes. Moreover, their use in an aqueous extraction process only promotes appearance and stabilization of the emulsions which they try to reduce.

Parameters of extraction

The enzymatic mix, the different parts of the process and the conditions under which the enzymatic attack is to be made are still not determined. This attack directly depends not only on pH, temperature, quantity of used enzymes and hydrolysis time but also on the treatments before or after kernels enzymatic hydrolysis. The efficiency of enzymes will not be the same on the seeds, the meal or for the fine crushed kernels. Factors influencing extraction by enzymatic way are:

- grinding: type of crusher, speed, humid or dry;
- inactivation of endogenous enzymes: equipments, pH, humidity, temperature, time;
- water quantity and mixture: water/solid ratio, equipments, speed rate, time;
- enzymatic hydrolysis: time, agitation, enzyme/substrate ratio, pH, temperature, sort of enzyme;
- inactivation of exogenous enzymes: temperature, pH, time, equipments;
- oil recovery: centrifugation, filtering, speed rate, time, pH, temperature.

It is also necessary to add to this list other treatments that the kernel, before enzymatic hydrolysis (pressing, for instance) or after enzymatic hydrolysis (pressing, extraction by solvent, etc.), can be subjected to.

Grinding: regarding the study of grinding in a-equi form, it is possible to cite Rosenthal et al. (1998) who put in an obvious place the effect the level of crushing has on the yield of extraction in oil and in protein; more thin is the crushing, more are the cell walls destroyed, and better are the yields. This study allows even to conclude that the effect of enzymes on soya is insignificant in comparison with the role of the particle size — without enzyme, the oil extraction yield resulting from an extraction in aqueous media ranged from 28% to 66% where the size of the particles of soya varied from 150 to 850 μm. The same results are obtained on germ corn (Dickey et al., 2008).

Endogenous/exogenous enzymes inactivation: it is necessary to preserve oil quality to disable a lot of endogenous enzymes such as myrosinase or lipase before the aqueous extraction. For myrosinase, Van Elyen et al. (2006) investigated the thermal inactivation kinetics of mustard seeds myrosinase in intact tissue: enzymatic activities are not detected at 72.5 °C after 30 min or at 75 °C after 10 min. The authors have shown that the time of treatment decreased when the seeds are not intact. Boiling intact rapeseed for 5 min can inactivate myrosinases effectively and 11.28% of glucosinolates is degraded due to the heating (Zhang and Wang, 2006). The inactivation by heating in an aqueous media is easier than in the whole seed (at 90 °C, the reference temperature). The authors have also shown that inactivation of rapsseed lipase is better with a microwave treatment than with a steam heating. And finally, the free fatty acid (FFA) content increases with the time of treatment but is the same in both cases (microwave vs. steam heating).

Water quantity and mixture: an optimum solid/water ratio was established by Sineiro et al. (1998) on sunflower at 7.5–8 g water per gram of seeds, whereas Hagenmaier (1974) worked with 1:10. For peanut oil extraction, solid:waters ratios recommended in the literature vary from 1:5 to 1:12. Embong and Jelen (1980) reported an optimum ratio varying between 1:2.5 and 1:3.5 for rapsseed oil extraction. In fact, the two different ways are opposite: for generating less effluent and less stable emulsion, it is necessary to have a high solid/water ratio in the extraction step. However, to obtain the highest extraction rates and extraction yields, it is usually necessary to use large quantities of water. The differences between these ratios come certainly from the wall cell composition, but also from different equipments of extraction and agitation.

Importance of enzymatic mixture

The enzymes, presented in the first part, were tested alone or in combination on numerous substrates, such as soya (Adler-Nissen, 1986; Yoon et al., 1991; Olsen, 1982; Bargale et al., 2000), corn (Boceska et al., 1993), olive (Fantozzi et al., 1977; Ranalli and Martinelli, 1994), coconut (McGlone et al., 1986; Christensen, 1989), avocado (Buenrostro and Lopez-Munguia, 1986; Freitas et al., 1993), sunflower (Lanzani et al., 1975; Dominguez et al., 1995, 1996), palm (Cheah et al., 1990), rice bran (Hanmounjai et al., 2000) and peanuts (Jiang et al., 2010). The results obtained on extraction yields vary. However, they all
come down to the same idea that the selection of multi-activity enzymes is always the most competitive, because as different polymers of the plant walls overlap, the effect of an enzyme will favour the release of a substrate of another enzyme (shown by Mathlouthi et al. (2002) on numerous substrates, including soybean, sunflower and rapeseed meals with xylanase and β-glucanase). So, the more important the number of enzymatic activities in media, the quicker and the more complete the hydrolysis of cell walls. However, any general implementation of an optimum mixture is impossible, as each species and each variety has different parietal compositions. The knowledge of the composition of the kernel or fruit cell walls must allow an enzymatic activity mixture to be composed in the same proportions in favouring a particular type of activity, because its substrate is in a bigger or smaller proportion.

Different studies on rapeseed about enzymatic mix have documented the results summarized in table 1.

Following Lanzani’s results, it was realized that the addition of pectinase could increase the yields to 4%. Sosulski also increased his yield by adding activities to a pectinolytic mixture.

Therefore, the number of enzymatic activities which are present in the mix is very important. All these factors have a real importance on enzymatic hydrolysis and free oil yield, and on oil and proteins quality as well as on the quantity of emulsion obtained after treatment. They can therefore produce some oil, but it is necessary to minimize emulsion, by supporting the quality of oil and of protein.

### Oil quality

Zhang et al. (2007) developed an interesting study of oil quality. Indeed, compared to solvent-extracted oil, enzyme-extracted oil contains more free fatty acids while its peroxide value is lower. The colour of enzyme-extracted oil is slightly darker than solvent-extracted oil’s. The iodine value, the saponification value and fatty acid composition are similar. Protein hydrolysates are composed of about 96% of peptides smaller than 1,500 Da, among about 87% are less than 600 Da. Protein hydrolysates are purified by macroporous adsorption resins. The sugar and ash contents of protein hydrolysates are significantly reduced and no glucosinolates or phytic acids are detected after purification.

### Emulsion

Emulsion is necessary to get a good yield in free oil. To control emulsion, parameters which favour the water/oil mix (quantity of water, agitation), as well as the parameters which support the stability of mixture (surfactant), are required. This is the case with proteases which produce peptides, and are well known for their amphiphilic property.

Guillemin (2006) made proteases act on rapeseed and has acquired yields close to those found by Zhang in 2007. However, the majority of oil is contained in the emulsion, and it has been proved that it is impossible to clear (with testing physical-chemist, thermodynamics or ultrasound way) the emulsion.

### Conclusion

Many researchers have worked on oil extraction in aqueous media, with and without enzymes. Many seeds or fruits have been studied over the last 50 years. Why this research lasts so long and why this work should be continued?

The research lasted so many years for a very simple reason. The process of extraction with hexane always provides the best returns. With more than 96% of oil recovered, these processes are almost perfect. Three important reasons can question the use of hexane:
- air pollutant;
- dangerousness;
- cost.

These three aspects make the research around this issue persists to this day. The alternative extraction techniques are improved or created regularly: the AEP, the supercritical CO₂ extraction, etc.; without ever reaching for the moment, the process returns to hexane extraction.

Thus, these processes remain incomplete and do not always grow at an industrial scale. Why research is still continued in this track?

Because these alternative procedures, including the AEP, are still promising. For further extraction of oil, which yields steady progress for years, we can notice that many publications in the area also speak of the extraction of proteins, which are recoverable by-products, while it is more difficult in the case of existing processes.

If high oil yields are reached and the extracted proteins are good, then the AEP may be an economically viable process of extraction. They will be industrially interesting, especially since the oil produced, if they do not achieve sufficient returns, have qualities superior to those obtained by processes using solvents. The conclusions of main publications on this subject show several points repeatedly:
- the specific substrates and enzymes used;
- the importance of enzymes and their conditions of employment (specific activities, duration, pH, temperature, ratio of seed/water ratio of enzyme/seed);

### Table 1. Yields of aqueous enzymatic process on rapeseed.

| Enzyme            | Temperature (°C) | Time (h) | Extraction yield | References               |
|-------------------|------------------|----------|------------------|--------------------------|
| Protease          | 40-50-65         | 3        | 74% total oil    | Lanzani et al., 1975     |
| Pectinase         | 40-50-60         | 3        | 78% total oil    | Lanzani et al., 1975     |
| a-amylose         | 50-63            | 3        | 72% total oil    | Fullbrook, 1982          |
| β-glucanase       |                  |          |                  |                          |
| Protease          |                  |          |                  |                          |
| Hemicellulase     | 50-53            | 3        | 75% total oil    | Fullbrook, 1982          |
| Multi-activity    | 50               | 12       | 43% dry matter   | Sosulski et al., 1988    |
| Pectinase         | 50               | 12       | 39% dry matter   | Sosulski et al., 1988    |
| Multi-activity    | 45-50            | 6        | Not determined   | Sosulski and Sosulski, 1990 |
| Multi-activity    | 50               | 4        | Not determined   | Olsen, 1988              |
| Multi-activity    | Not determined   | Not determined | 80% total oil  | Deng et al., 1992        |
| Pectinase         | 48               | 3        | 83% total oil    | Zhang et al., 2007       |
| Cellulase β-glucanase | 48            | 5        | 88-90% total oil | Zhang and Wang, 2006     |

References

- C) Time (h) Extraction
- Yields of aqueous enzymatic process on rapeseed.
- Emulsion
- Protease
- α-amylase
- β-glucanase
- Hemicellulase
- Multi-activity
- Pectinase
- Cellulase
- β-glucanase
- Protein hydrolysates are composed of about 96% of peptides smaller than 1,500 Da, among about 87% are less than 600 Da. Protein hydrolysates are purified by macroporous adsorption resins. The sugar and ash contents of protein hydrolysates are significantly reduced and no glucosinolates or phytic acids are detected after purification.

Conclusion

Many researchers have worked on oil extraction in aqueous media, with and without enzymes. Many seeds or fruits have been studied over the last 50 years. Why this research lasts so long and why this work should be continued?

The research lasted so many years for a very simple reason. The process of extraction with hexane always provides the best returns. With more than 96% of oil recovered, these processes are almost perfect. Three important reasons can question the use of hexane:
- air pollutant;
- dangerousness;
- cost.

These three aspects make the research around this issue persists to this day. The alternative extraction techniques are improved or created regularly: the AEP, the supercritical CO₂ extraction, etc.; without ever reaching for the moment, the process returns to hexane extraction.

Thus, these processes remain incomplete and do not always grow at an industrial scale. Why research is still continued in this track?

Because these alternative procedures, including the AEP, are still promising. For further extraction of oil, which yields steady progress for years, we can notice that many publications in the area also speak of the extraction of proteins, which are recoverable by-products, while it is more difficult in the case of existing processes.

If high oil yields are reached and the extracted proteins are good, then the AEP may be an economically viable process of extraction. They will be industrially interesting, especially since the oil produced, if they do not achieve sufficient returns, have qualities superior to those obtained by processes using solvents. The conclusions of main publications on this subject show several points repeatedly:
- the specific substrates and enzymes used;
- the importance of enzymes and their conditions of employment (specific activities, duration, pH, temperature, ratio of seed/water ratio of enzyme/seed);

References
– the pre-hydrolysis treatment (cleaning, grinding, inactivation of endogenous enzymes);
– and separation techniques.
By setting these factors, and studying their influence and importance on the AEP, it is probably possible to achieve economically attractive yields.

REFERENCES

Adler-Nissen J. A review of food protein hydrolysis-specific areas. In: Adler-Nissen, ed. Enzymic Hydrolysis of Food Proteins. London and New York: Elsevier Applied Science Publishers, 1986.

Bargale PC, Sosulski K, Sosulski FW. Enzymatic hydrolysis of corn germ oil obtained by aqueous enzymatic extraction. J Food Process Engineer 2000; 23: 321-7.

Cheah SC, Augustin MA, Ooi LCL. Enzymic extraction of avocado oil.

Dominguez H, Nunez M, Lema J. Enzymatic pretreatment to enhance oil extraction from fruits and oilseeds: a review. Food Chem 1994; 49: 271-86.

Freitas SP, Lago RCA, Jablonka FH, Hartman L. Aqueous enzymatic extraction of avocado oil from fresh pulp. Rev Fr Corps Gras 1993; 40: 365-71.

Fullbrook PD. The use of enzymes in the processing of oilseeds. J Am Oil Chem Soc 1982; 60: 476-8.

Guillem S. Extraction aqueuse d’huile de colza assistée par hydrolyse enzymatique: optimisation de la réaction, caractérisation de l’émulsion et étude des procédés de destabilisation. Thèse de l’Institut National Polytechnique de Lorraine, 2006.

Hagenmaier R. Aqueous processing of full-fat sunflower seeds: Yields of oil and protein. J Am Oil Chem Soc 1974; 51: 470-1.

Hammond P, Pyle L, Niranjan K. Extraction of rice bran oil using aqueous media. J Chem Tech Biot 2000; 75: 348-52.

Jiang L, Hua D, Wang Z, Xu S. Aqueous enzymatic extraction of peanut oil and protein hydrolysates. Food Bioprod Process 2010; 88: 233-8.

Johnson LA, Lucas W. Comparison of alternative solvents for oils extraction. J Am Oil Chem Soc 1983; 60: 229-42.

Lanzani A, Petrini M, Cozzoli O, Gallavresi P, Carola C, Jaconi G. On the use of enzymes for vegetable oil extraction. A preliminary report. Riv Ital Sostanze Grasse 1975; 11: 226-9.

Matthloutti N, Saulnier L, Quemener B, Labrier M. Xylanase, β-glucanase, and other side enzymatic activities have greater effects on the viscosity of several feedstocks than xylanase and β-glucanase used alone or in combination. J Agr Food Chem 2002; 50: 5121-7.