Synthesis and Biological activity of 4-(4,6-Disubstituted-pyrimidin-2-yloxy)phenoxy Acetates

Lin Jiang 1,*, Hao Wang 1, Maorong Wang 1 and Xinhuan Teng 2

1 College of Chemistry and Material Science, Shandong Agricultural University, Taian 271018, China; E-Mails: hw@sdau.edu.cn (H.W.), maorongwang@163.com (M.R.W.)

2 College of Plant Protection, Shandong Agricultural University, Taian 271018, China; E-Mail: tengxinhuan_0505@163.com (X.H.T.)

* Author to whom correspondence should be addressed; E-mail: jiangl@sdau.edu.cn

Received: 14 January 2010; in revised form: 1 February 2010 / Accepted: 20 February 2010 / Published: 23 February 2010

Abstract: Ten novel 4-(4,6-dimethoxypyrimidin-2-yloxy)phenoxy acetates and 4-(4,6-dimethylpyrimidin-2-yloxy)phenoxy acetates were synthesized with hydroquinone, 2-methylsulfonyl-4,6-disubstituted-pyrimidine and chloroacetic ester as starting materials. The products were characterized by IR, 1H-NMR, MS spectra and elemental analyses. Preliminary bioassay indicates that the target compounds possess high herbicidal activity against monocotyledonous plants such as Digitaria sanguinalis L. at concentrations of 100 mg/L and 50 mg/L.

Keywords: pyrimidine; phenoxyacetate; synthesis; herbicidal activity

1. Introduction

Aryloxy-phenoxy propionates are an important class of herbicides due to their high efficiency, broad spectrum, low toxicity and good selectivity. They act by blocking the biosynthesis of fatty acids by inhibiting acetyl-coenzyme A carboxylase [1–3]. Since the first herbicide of this series, diclofop-methyl, was synthesized in 1972, more than twenty aryloxy-phenoxy propionate herbicides such as fluazifop-butyl, heloxyfop-methyl, quizalofop-ethyl and cyhalofop-butyl have been developed [4], and are widely used to control gramineous weeds. In addition, some aryloxy-phenoxy acetates exhibit good herbicidal activity. For example, two substituted pyrazolo[3,4-d] pyrimidin-4-yloxy phenoxy acetates
display considerable activities [5], with 100% inhibition against the root growth of *Brassica napus* L. at a concentration of 100 mg/L, and 98.1% and 100% against the root growth of *Echinochloa crusgalli* L. at the same concentration, respectively.

On the other hand, a lot of compounds containing 4,6-dimethoxypyrimidin-2-yl or 4,6-dimethylpyrimidin-2-yl moieties display excellent herbicidal activity. Most sulfonylurea herbicides and all pyrimidinylbenzoate herbicides (two series of super-efficient herbicides) [6–10], such as nicofuluron, amidosulfuron, halopyrazosulfuron, ethoxysulfuron, pyriminobac-methyl and pyrfluid, possess 4,6-dimethoxypyrimidin-2-yl groups, while sulfometuron-methyl, a kind of sulfonylurea, contains a 4,6-dimethylpyrimidin-2-yl group, which suggests that the two disubstituted-pyrimidin-2-yl groups possess high biological activity [11]. In order to seek efficient herbicidal compounds, herein we introduced 4,6-dimethoxypyrimidin-2-yl or 4,6-dimethylpyrimidin-2-yl into aryloxy-phenoxy acetates by the sub-structure link route, synthesizing ten novel 4-(4,6-dimethoxypyrimidin-2-yloxy)phenoxy acetates and 4-(4,6-dimethylpyrimidin-2-yloxy)phenoxy acetates.

2. Results and Discussion

2.1. Synthesis

There are two pathways (route A and route B, Scheme 1) in the synthesis of aryloxy-phenoxy propionate herbicides [12]. In route A, hydroquinone reacts with 2-chloropropionates to yield 4-hydroxyphenoxy propionates which react with aryl chlorides to give the target compounds. In route B, hydroquinone firstly reacts with an aryl chloride to give a 4-hydroxyphenyl aryl ether which reacts with 2-chloropropionates to give the target compound.

**Scheme 1.** Synthetic pathways of aryloxy-phenoxy propionates.

Considering the lower yield of 4-hydroxyphenoxy propionate in the reaction of hydroquinone with 2-chloropropionate in route A, we choose route B as the synthetic strategy in our work. Moreover, we used 2-methylsulfonyl-4,6-disubstitutedpyrimidines instead of 2-chloro-4,6-disubstitutedpyrimidines to produce 2-(4-hydroxyphenoxy)-4,6-disubstitutedpyrimidines 2a, 2b owing to their higher activity [13]. The reaction of hydroquinone with 2-methylsulfonyl-4,6-dimethoxypyrimidine in tetrahydrofuran or *N,N*-dimethylformamide at 70–80 °C [14] gives 2a in very low yield and low purity. Hence, depending on the preparation method of 2-(4-hydroxyphenoxy)-6-chlorobenzoxazole [15], we used sodium
hydroxide as a base, and benzyltriethylammonium chloride as a phase transfer catalyst, thus, refluxing the hydroquinone with 2-methylsulfonyl-4,6-dimethoxy pyrimidine for 3.5 h with stirring in a mixed solvent of toluene and water affords compound 2a with a satisfactory yield. The reaction of 2a (or 2b) with the appropriate chloroacetates in acetonitrile in the presence of potassium carbonate gives compounds 3a–3j (Scheme 2).

**Scheme 2.** Synthetic route of target compounds 3a–3j.

![Scheme 2](image)

|   | R<sub>1</sub> | R<sub>2</sub> |   | R<sub>1</sub> | R<sub>2</sub> |
|---|-------------|-------------|---|-------------|-------------|
| a | OCH<sub>3</sub> | CH<sub>3</sub> | f | CH<sub>3</sub> | CH<sub>3</sub> |
| b | OCH<sub>3</sub> | CH<sub>2</sub>CH<sub>3</sub> | g | CH<sub>3</sub> | CH<sub>2</sub>CH<sub>3</sub> |
| c | OCH<sub>3</sub> | CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> | h | CH<sub>3</sub> | CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> |
| d | OCH<sub>3</sub> | CH<sub>(CH<sub>3</sub>)</sub> | i | CH<sub>3</sub> | CH<sub>(CH<sub>3</sub>)</sub> |
| e | OCH<sub>3</sub> | CH<sub>(CH<sub>2</sub>)</sub>CH<sub>3</sub> | j | CH<sub>3</sub> | CH<sub>(CH<sub>2</sub>)</sub>CH<sub>3</sub> |

Both spectral and elemental analyses data of the prepared compounds 3a–3j were all in agreement with the suggested molecular structures (see Experimental). Their IR spectra exhibited absorption bands assignable to C=O stretching vibrations and characteristic bands near 1208–1193 cm<sup>-1</sup> and 1084–1062 cm<sup>-1</sup>, corresponding to the C-O-C linkages. In addition, the <sup>1</sup>H-NMR spectra of 3a–3j showed characteristic singlet signals at δ 4.59–4.64, 3.82–3.83, and 2.39 ppm, assignable to the methylene, methoxy and methyl protons, respectively, and singlet signals at δ 5.76 ppm or at δ 6.74–6.75 ppm corresponding to the dimethoxy- or dimethylpyrimidine ring protons (5-H), in addition to the expected aromatic protons appear as double doublet signals at 6.91–6.93 and 7.12–7.14 ppm, respectively.

### 2.2. Biological Activity

The herbicidal activities are summarized in Table 1. In all tested compounds, the rates of inhibition of *Brassica napus* L. root growth are 15.9%–58.4%, 13.1%–36.4% and 9.2%–22.9% at concentrations of 100 mg/L, 50 mg/L and 10 mg/L, respectively, which means that the synthesized compounds display low herbicidal activities against this plant. However, the rates of inhibition of *Digitaria sanguinalis* L. root growth are all 100% at a concentration of 100 mg/L, 95.2%–98.5% at 50 mg/L, and 32.5%–44.7% at 10 mg/L. The results demonstrate that the target compounds exhibit excellent herbicidal activities against this kind of weed at 100 mg/L and 50 mg/L, but their activities are not satisfactory at a lower concentration (10 mg/L).
Table 1. Herbicidal activities of compounds 3a–3j.

| Compd. | Inhibitory rate (%) | Brassica napus L. | Echinochloa crusgalli L. |
|--------|---------------------|------------------|-------------------------|
|        | 100 mg/L 50 mg/L 10 mg/L | 100 mg/L 50 mg/L 10 mg/L |
| 3a     | 39.8 34.4 22.9 | 100 | 98.5 | 42.8 |
| 3b     | 41.1 31.0 13.4 | 100 | 96.2 | 39.9 |
| 3c     | 15.9 13.1 10.1 | 100 | 95.2 | 32.5 |
| 3d     | 25.1 21.0 14.6 | 100 | 96.5 | 32.7 |
| 3e     | 28.3 13.4 9.5  | 100 | 96.9 | 41.8 |
| 3f     | 44.7 26.7 11.9 | 100 | 97.3 | 44.7 |
| 3g     | 58.4 32.7 14.2 | 100 | 98.4 | 38.5 |
| 3h     | 54.5 33.9 12.1 | 100 | 97.5 | 36.9 |
| 3i     | 26.4 14.3 9.2  | 100 | 95.1 | 35.0 |
| 3j     | 50.2 34.2 21.5 | 100 | 97.6 | 42.6 |
| quizalofop-P-ethyl | 51.2 39.2 33.8 | 100 | 95.3 | 63.5 |

3. Experimental

3.1. General

Melting points were measured on an X-5 microscopic melting-point apparatus and uncorrected. IR spectra were recorded in KBr pellets on a Shimadzu IR-440 infrared spectrophotometer. $^1$H-NMR spectra were registered on an Inova-600 spectrometer (in CDCl$_3$ solvent, TMS as internal standard). Mass spectra were recorded on an Agilent 1100 LC-MS spectrometer (APCI source). Elemental analyses were performed with a Vario EL III Elemental Analyzer.

3.2. General method for the synthesis of 2-(4-hydroxyphenoxy)-4,6-disubstitutedpyrimidines 2a, 2b

To a solution of sodium hydroxide (1.20 g, 30.0 mmol) and benzyltriethylammonium chloride (0.10 g, 0.4 mmol) in water (15 mL) were added hydroquinone (2.98 g, 27.0 mmol) and toluene (7 mL). The resulting mixture was heated to 60 °C while stirring under nitrogen, then 2-methylsulfonyl-4,6-disubstitutedpyrimidine 1a or 1b (18.0 mmol) in dichloromethane (8 mL) was added dropwise over 1 h. The reaction mixture was further refluxed with stirring for 3 h, then toluene and dichloromethane were evaporated in a rotary evaporator. Sodium hydroxide (5%, 30 mL) was added to the residue, and extracted with ethyl acetate (10 mL × 2). The aqueous layer was acidified with concentrated hydrochloric acid to pH4–5. The precipitated solid was filtered off, washed with water, dried and crystallized from a mixed solvent of ethanol and water (V:V = 1:2) to give 2a, 2b respectively.

2-(4-Hydroxyphenoxy)-4,6-dimethoxypyrimidine (2a): Yield: 86.5%; m.p. 120.1-121.8 °C; $^1$H-NMR δ: 3.84 (s, 6H, CH$_3$), 5.76 (s, 1H, Pyrim-H), 5.79 (s, 1H, OH), 6.82 (d, 2H, $J = 8.4$ Hz, Ph-H), 7.03 (d, 2H, $J = 8.4$ Hz, Ph-H) ppm; IR (KBr): 3453, 1620, 1573, 1513, 1454, 812 cm$^{-1}$; MS (APCI) m/z: 249.9 (M + 2H)$^+$; Anal. for C$_{12}$H$_{12}$N$_2$O$_4$: (%) C 58.06, H 4.87, N 11.28; found C 58.31, H 4.74, N 11.41.
2-(4-Hydroxyphenoxy)-4,6-dimethylpyrimidine (2b): Yield: 90.6%, m.p. 195.8-197.2 °C; \(^1\)H-NMR \(\delta\): 2.41 (s, 6H, CH\(_3\)), 6.76 (s, 1H, Pyrim-H), 6.78 (d, 2H, \(J = 8.4\) Hz, Ph-H), 6.98 (d, 2H, \(J = 8.4\) Hz, Ph-H), 7.45 (s, 1H, OH) ppm; IR (KBr): 3448, 1637, 1560, 1509, 1458, 820 cm\(^{-1}\); MS (APCI) m/z: 218.0 (M + 2H)\(^+\); Anal. for C\(_{12}\)H\(_{12}\)N\(_2\)O\(_2\) (%): C 66.65, H 6.66, N 12.95; found C 66.69, H 6.21, N 12.91.

3.3. General method for the preparation of 4-(4,6-disubstitutedpyrimidin-2-yloxy)phenoxy acetates 3a–3j

To a mixture of 2-(4-hydroxyphenoxy)-4,6-disubstitutedpyrimidine 2a or 2b (5.0 mmol) and anhydrous potassium carbonate (0.97 g, 7.0 mmol) in acetonitrile (10 mL), was added the appropriate chloroacetic ester (6.0 mmol) while stirring. The reaction mixture was refluxed with stirring for 7 h. The solvent was removed in a rotary evaporator, then, the residue was poured into ice-water (20 mL), and leave aside for 1 h to solidify. The solid so formed was filtered off, washed with ice-water, dried and crystallized from a mixed solvent of petroleum ether and ethyl acetate (V: V = 5:1–2) to afford 3a–3j.

Methyl 4-(4,6-dimethoxypyrimidin-2-yloxy)phenoxy acetate (3a): Yield: 81.2%; m.p. 77.5–79.0 °C; \(^1\)H-NMR \(\delta\): 3.82 (s, 3H, COOCH\(_3\)), 3.83 (s, 6H, Pyrim-OCH\(_3\)), 4.64 (s, 2H, OCH\(_2\)), 5.76 (s, 1H, Pyrim-H), 5.76 (s, 2H, J = 9.0 Hz, Ph-H), 7.14 (d, 2H, \(J = 9.0\) Hz, Ph-H) ppm; IR (KBr): 1754, 1205, 1083, 820 cm\(^{-1}\); MS (APCI) m/z: 321.9 (M + 2H)\(^+\); Anal. for C\(_{15}\)H\(_{16}\)N\(_2\)O\(_6\) (%): C 56.25, H 5.04, N 8.74; found C 56.51, H 4.99, N 8.79.

Ethyl 4-(4,6-dimethoxypyrimidin-2-yloxy)phenoxy acetate (3b): Yield: 78.5%; m.p. 75.2–76.1 °C; \(^1\)H-NMR \(\delta\): 1.30 (t, 3H, \(J = 7.2\) Hz, CH\(_2\)-CH\(_3\)), 3.83 (s, 6H, Pyrim-OCH\(_3\)), 4.28 (q, 2H, \(J = 7.2\) Hz, COOCH\(_2\)), 4.62 (s, 2H, OCH\(_2\)), 5.76 (s, 1H, Pyrim-H), 5.76 (d, 2H, J = 9.0 Hz, Ph-H), 7.14 (d, 2H, \(J = 9.0\) Hz, Ph-H) ppm; IR (KBr): 1758, 1205, 1068, 825 cm\(^{-1}\); MS (APCI) m/z: 335.9 (M + 2H)\(^+\); Anal. for C\(_{16}\)H\(_{18}\)N\(_2\)O\(_6\) (%): C 57.48, H 5.43, N 8.38; found C 57.75, H 5.35, N 8.25.

Propyl 4-(4,6-dimethoxypyrimidin-2-yloxy)phenoxy acetate (3c): Yield: 75.6%; m.p. 82.8–84.3 °C; \(^1\)H-NMR \(\delta\): 0.93 (t, 3H, \(J = 7.2\) Hz, CH\(_2\)-CH\(_3\)), 1.69 (m, 2H, CH\(_2\)-CH\(_3\)), 3.83 (s, 6H, Pyrim-OCH\(_3\)), 4.18 (t, \(J = 6.6\) Hz, 2H, COOCH\(_2\)), 4.63 (s, 2H, OCH\(_2\)), 5.76 (s, 1H, Pyrim-H), 5.76 (d, 2H, J = 9.0 Hz, Ph-H), 7.14 (d, 2H, \(J = 9.0\) Hz, Ph-H) ppm; IR (KBr) 1759, 1206, 1066, 829 cm\(^{-1}\); MS (APCI) m/z: 350.0 (M + 2H)\(^+\); Anal. for C\(_{17}\)H\(_{20}\)N\(_2\)O\(_6\) (%): C 58.61, H 5.79, N 8.04; found C 58.71, H 5.72, N 8.05.

Isopropyl 4-(4,6-dimethoxypyrimidin-2-yloxy)phenoxy acetate (3d): Yield: 80.9%; m.p. 79.8–81.5 °C; \(^1\)H-NMR \(\delta\): 0.93 (t, 3H, \(J = 6.6\) Hz, CH\(_2\)-(CH\(_3\))\(_2\)), 1.69 (m, 2H, CH\(_2\)-CH\(_3\)), 3.83 (s, 6H, Pyrim-OCH\(_3\)), 4.59 (s, 2H, OCH\(_2\)), 5.15 (m, 1H, COOCH), 5.76 (s, 1H, Pyrim-H), 6.91 (d, 2H, \(J = 9.0\) Hz, Ph-H), 7.12 (d, 2H, J = 9.0 Hz, Ph-H) ppm; IR (KBr): 1760, 1213, 1062, 815 cm\(^{-1}\); MS (APCI) m/z: 349.9 (M + 2H)\(^+\); Anal. for C\(_{17}\)H\(_{20}\)N\(_2\)O\(_6\) (%): C 58.61, H 5.79, N 8.04; found C 58.65, H 5.71, N 8.02.

Butyl 4-(4,6-dimethoxypyrimidin-2-yloxy)phenoxy acetate (3e): Yield: 71.3%; m.p. 70.1–71.3 °C; \(^1\)H-NMR \(\delta\): 0.93 (t, 3H, \(J = 7.2\) Hz, CH\(_2\)-CH\(_3\)), 1.36 (m, 2H, CH\(_2\)-CH\(_3\)), 1.64 (m, 2H, CH\(_2\)-C\(_2\)H\(_5\)), 3.83 (s, 6H, Pyrim-OCH\(_3\)), 4.22 (t, 2H, \(J = 6.6\) Hz, COOCH\(_2\)), 4.63 (s, 2H, OCH\(_2\)), 5.76 (s,
1H, Pyrim-H), 6.91 (d, 2H, \( J = 9.0 \) Hz, Ph-H), 7.14 (d, 2H, \( J = 9.0 \) Hz, Ph-H) ppm; IR (KBr): 1763, 1208, 809 cm\(^{-1}\); MS (APCI) m/z: 363.9 (M + 2H)\(^+\); Anal. for C\(_{18}\)H\(_{22}\)N\(_2\)O\(_6\) (%): C 59.66, H 6.12, N 7.73; found C 59.49, H 6.08, N 7.77.

Methyl 4-(4,6-dimethylpyrimidin-2-yloxy)phenoxy acetate (3f): Yield: 80.1%; m.p. 114.5–116.0 ºC; 1H-NMR \( \delta \): 2.39 (s, 6H, Pyrim-CH\(_3\)), 3.82 (s, 3H, COOCH\(_3\)), 4.64 (s, 2H, OCH\(_2\)), 6.75 (s, 1H, Pyrim-H), 6.92 (d, 2H, \( J = 9.0 \) Hz, Ph-H), 7.13 (d, 2H, \( J = 9.0 \) Hz, Ph-H) ppm; IR (KBr): 1753, 1200, 1082, 833 cm\(^{-1}\). MS (APCI) m/z: 289.9 (M + 2H)\(^+\); Anal. for C\(_{15}\)H\(_{16}\)N\(_2\)O\(_4\) (%): C 62.53, H 5.59, N 9.71; found C 62.86, H 5.55, N 9.70.

Ethyl 4-(4,6-dimethylpyrimidin-2-yloxy)phenoxy acetate (3g): Yield: 77.2%; m.p. 90.2–92.1 ºC; 1H-NMR \( \delta \): 1.31 (t, 3H, \( J = 7.2 \) Hz, CH\(_2\)-CH\(_3\)), 2.39 (s, 6H, Pyrim-CH\(_3\)), 4.28 (q, 2H, \( J = 7.2 \) Hz, COOCH\(_2\)), 4.62 (s, 2H, OCH\(_2\)), 6.74 (s, 1H, Pyrim-H), 6.92 (d, 2H, \( J = 9.0 \) Hz, Ph-H), 7.13 (d, 2H, \( J = 9.0 \) Hz, Ph-H) ppm; IR (KBr): 1758, 1197, 1078, 845 cm\(^{-1}\). MS (APCI) m/z: 303.9 (M + 2H)\(^+\); Anal. for C\(_{16}\)H\(_{18}\)N\(_2\)O\(_4\) (%): C 63.62, H 6.01, N 9.26; found C 64.03, H 5.98, N 9.33.

Propyl 4-(4,6-dimethylpyrimidin-2-yloxy)phenoxy acetate (3h): Yield: 75.0%; m.p. 95.4–97.3 ºC; 1H-NMR \( \delta \): 0.93 (t, 3H, \( J = 7.2 \) Hz, CH\(_2\)-CH\(_3\)), 1.69 (m, 2H, CH\(_2\)-CH\(_3\)), 2.39 (s, 6H, Pyrim-CH\(_3\)), 4.18 (t, \( J = 6.6 \) Hz, 2H, COOCH\(_2\)), 6.74 (s, 1H, Pyrim-H), 6.93 (d, 2H, \( J = 9.0 \) Hz, Ph-H), 7.13 (d, 2H, \( J = 9.0 \) Hz, Ph-H) ppm; IR (KBr): 1760, 1198, 1078, 849 cm\(^{-1}\). MS (APCI) m/z: 317.9 (M + 2H)\(^+\); Anal. for C\(_{17}\)H\(_{20}\)N\(_2\)O\(_4\) (%): C 64.54, H 6.37, N 8.85; found C 64.84, H 6.28, N 8.76.

Isopropyl 4-(4,6-dimethylpyrimidin-2-yloxy)phenoxy acetate (3i): Yield: 80.9%; m.p. 90.5–92.2 ºC; 1H-NMR \( \delta \): 1.28 (d, 6H, \( J = 6.6 \) Hz, CH-(CH\(_3\))\(_2\)), 2.39 (s, 6H, Pyrim-CH\(_3\)), 4.59 (s, 2H, OCH\(_2\)), 5.15 (m, 1H, COOCH), 6.74 (s, 1H, Pyrim-H), 6.92 (d, 2H, \( J = 9.0 \) Hz, Ph-H), 7.12 (d, 2H, \( J = 9.0 \) Hz, Ph-H) ppm; IR (KBr): 1721, 1204, 1073, 854 cm\(^{-1}\). MS (APCI) m/z: 317.9 (M + 2H)\(^+\); Anal. for C\(_{17}\)H\(_{20}\)N\(_2\)O\(_4\) (%): C 64.54, H 6.37, N 8.85; found C 64.84, H 6.28, N 8.76.

Butyl 4-(4,6-dimethylpyrimidin-2-yloxy)phenoxy acetate (3j): Yield: 71.3%; m.p. 34.2–36.0 ºC; 1H-NMR \( \delta \): 0.92 (t, 3H, \( J = 7.2 \) Hz, CH\(_2\)-CH\(_3\)), 1.37 (m, 2H, CH\(_2\)-CH\(_3\)), 1.65 (m, 2H, CH\(_2\)-C\(_2\)H\(_5\)), 2.39 (s, 6H, Pyrim-CH\(_3\)), 4.22 (t, 2H, \( J = 6.6 \) Hz, COOCH\(_2\)), 4.63 (s, 2H, OCH\(_2\)), 6.74 (s, 1H, Pyrim-H), 6.92 (d, 2H, \( J = 9.0 \) Hz, Ph-H), 7.13 (d, 2H, \( J = 9.0 \) Hz, Ph-H) ppm; IR (KBr): 1752, 1193, 1084, 830 cm\(^{-1}\). MS (APCI) m/z: 331.9 (M + 2H)\(^+\); Anal. for C\(_{18}\)H\(_{22}\)N\(_2\)O\(_4\) (%): C 65.44, H 6.71, N 8.48; found C 65.14, H 6.74, N 8.21.

3.4. Biological activity

The herbicidal activities of the target compounds were determined using Brassica napus L. and Digitaria sanguinalis L. as samples of dicotyledonous and monocotyledonous plants, respectively [16,17]. Emulsions of the tested compounds were prepared by dissolving them in N,N-dimethylformamide (100 μL) with the addition of Tween 20 (2 μL), and then diluting with distilled water. The germinated seeds were placed on two filter papers in a 9-cm Petri plate, to which 5 mL of tested solution was added in advance. Usually, 15 seeds were used on each plate. The plates
were placed in a dark room and allowed to germinate for 65 h at 28 (±1) °C. The lengths of 10 seed roots selected from each plate were measured and the means were calculated. Moreover, quizalofop-P-ethyl, a commercial aryloxy-phenoxy propionate herbicide and the emulsion which does not contain tested compounds were used as control and blank respectively. For all of the bioassay tests, each treatment was repeated three times. The inhibitory rate was calculated relative to the blank. The bioassay results are shown in Table 1.

4. Conclusions

A new method of preparing 2-(4-hydroxyphenyloxy)-4,6-disustedpyrimidine has been developed with the advantages of a shorter reaction time, fewer side-products and higher yields. Ten novel 4-(4,6-dimethoxypyrimidin-2-yloxy)phenoxy acetate and 4-(4,6-dimethylpyrimidin-2-yloxy)phenoxy acetate compounds have been synthesized, and characterized by IR, 1H-NMR, MS spectra and elemental analyses. Preliminary bioassay indicates that the target compounds exhibit high herbicidal activity against monocotyledonous plant such as *Digitaria sanguinalis* L. at concentrations of 100 mg/L and 50 mg/L.

Acknowledgements

The authors are grateful to Prof. Wang Jin-Xin of Shandong Agricultural University for the herbicidal activity tests.

References and Notes

1. Ren, K.T.; Li, Y.H.; Yang, H.Z. The mechanisms of action of APP and CHD herbicides. *Chin. J. Pesticides* 1999, 38, 1–4.
2. Zhu, X.L.; Zhang, L.; Chen, Q.; Wan, J.; Yang, G.F. Interactions of aryloxyphenoxypropionic acids with sensitive and resistant acetyl-coenzyme A carboxylase by homology modeling and molecular dynamic simulations. *J. Chem. Inf. Model.* 2006, 46, 1819–1826.
3. Zhu, X.L.; Hao, G.F.; Zhan, C.G.; Yang, G.F. Computational simulations of the interactions between acetyl-coenzyme-A carboxylase and clodinafop: Resistance mechanism due to active and nonactive site mutations. *J. Chem. Inf. Model.* 2009, 49, 1936–1943.
4. Li, H.P. Developing progress and characteristics of herbicides of novel varieties in the world from 1980s. *Pestic. Sci. Adm.* 2004, 25, 28–32.
5. Wang, H.Q.; Liu, H.; Liu, Z.J. Synthesis and herbicidal activities of 1-phenyl-1,5-dihydro-2-methylthio-methylsulfonyl-5-aryl-6-(4-ethyloxycarbonylmethoxyphenoxy)-pyrazolo[3,4-d] pyrimidin-4-one. *Chin. J. Org. Chem.* 2004, 24, 1563–1568.
6. Liu, C.L. Recent advance on the development of sulfonylurea herbicide. *Pestic. Sci. Adm.* 2000, 21, 35–39.
7. Xi, Z.; Yu, Z.H.; Niu C.W.; Ban, S.R.; Yang, G.F. Development of a general quantum-chemical descriptor for steric effects: density functional theory based QSAR study of herbicidal sulfonylurea analogues. *J. Comput. Chem.* 2006, 27, 1572–1576.
8. He, Y.Z.; Li, Y.X.; Zhu, X.L.; Xi, Z.; Niu, C.W.; Wan, J.; Zhang, L.; Yang, G.F. Rational design based on bioactive conformation analysis of pyrimidinylbenzoates as acetohydroxyacid synthase inhibitors by integrating molecular docking, CoMFA, CoMSIA, and DFT calculations. *J. Chem. Inf. Model.* **2007**, *47*, 2335–2344.

9. Li, Y.X.; Luo, Y.P.; Xi, Z.; Niu, C.W.; He, Y.Z.; Yang, G.F. Design and synthoses of novel phthalazin-1(2H)-one derivatives as acetohydroxyacid synthase inhibitors. *J. Agric. Food Chem.* **2006**, *54*, 9135–9139.

10. Tamaru, M.; Takehi, T.; Masuyama, N.; Hanai, R. Studies of the new herbicide KIH-6127. Part II. Synthesis and herbicidal activity of 6-acyl pyrimidin-2-yl salicylates and analogues against barnyard grass. *Pestic. Sci.* **1996**, *47*, 327–335.

11. Gerorge, L. Agricultural sulfonamides. U.S. Patent 4,394,506, 1983.

12. Tang, C.C.; Li, Y.C.; Chen, B.; Yang, H.Z.; Jin, G.Y. *Pesticide Chemistry*; Nankai University Press: Tianjin, China, 1998; p. 558.

13. Fan, Q.J.; Shen, D.L.; Chen, J.Q. Synthesis and application research of 4,6-dimethoxy-2-methylsulfonylpyrimidine. *Fine Chem. Intermed.* **2005**, *35*, 39–41.

14. Wu, J.; Cheng, J.; Lu, L. N-(2-Bromophenyl)-2-(4,6-dimethoxypyrimidin-2-yl oxy)benzylamine, a new selective postemergent herbicide for weed control in winter oilseed rape. *J. Agric. Food Chem.* **2006**, *54*, 5954–5957.

15. Wu, Y.H. The new synthesis method of 2-(4-hydroxyphenoxy)-6-chlorobenzoxazole. *Chin. J. Anhui Huagong* **2005**, *5*, 22–23.

16. Mu, L.Y. *Research Method of Plant Chemical Protection*; China Agriculture Press: Beijing, China, 1994; pp. 90–91.

17. Wang, B.L.; Duggleby, R.G.; Li, Z.M.; Wang, J.G.; Li, Y.H.; Wang, S.H.; Song, H.B. Synthesis, crystal structure and herbicidal activity of mimics of intermediates of the KARI reaction. *Pest Manag. Sci.* **2005**, *61*, 407–412.

*Sample Availability:* Samples of the target compounds are available from the authors.

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