Polycyclic Aromatic Hydrocarbons and Selected Heavy Metals in Some Oil Polluted Sites in Delta State Nigeria

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Abstract

Concentrations of selected heavy metals, nutrient elements and PAHs in farms and produce (cassava tubers and oil bean seeds) from 4-year-old crude oil impacted areas (Ekore and Uduvwoku) and a non-oil-impacted area (Okpe), all in Ughelli South Local Government Area, Delta State, Nigeria, were investigated to ascertain degree of risk posed. A random sampling design was chosen with three replications. Results obtained revealed significantly (P < 0.05) elevated (mg·kg⁻¹) Cd (0.240, 0.140) and Cr (1.327, 3.122) in cassava samples for Ekore and Uduvwoku respectively in comparison to non-detectable amount for those of non-impacted source and exceeded set WHO limits of 0.1 and 0.05 mg·kg⁻¹ respectively. Although PAHs were low, oil spill increased available levels by factor of 2.5 and 5 for Ekore and Uduvwoku respectively in comparison to non-detectable amount for those of non-impacted source and exceeded set WHO limits of 0.1 and 0.05 mg·kg⁻¹ respectively. Although PAHs were low, oil spill increased available levels by factor of 2.5 and 5 for Ekore and Uduvwoku respectively in comparison to non-detectable amount for those of non-impacted source and exceed set WHO limit for study impacted soils. Available N, P and K decreased (%) by 56.1, 28.5 and 2.4 for Ekore and 82.9, 39.9 and 45.5 for Uduvwoku Cassava samples. Nutrient profiling in oil bean revealed % reduction in available N, P and K by 33.7, 47.7 and 57.9 and 28.9, 76.3 and 39.8 for Ekore and Uduvwoku samples respectively. For oil bean, Cd and Cr did not differ markedly between polluted samples but exceeded WHO limits in both soils. Other studied contaminants fell within limits. In soils, available N, P and K decreased (%) by 39.6, 79.1 and 27.4 for Ekore and 53, 88.1 and 45.5 for Uduvwoku samples. Low pH of 5.3 and 5.7 in Ekore and Uduvwoku respectively may increase the leachability of Cr into groundwater. Biopersistent Cd and Cr were found to biomagnify up the food chain and may impair major processes. Although PAHs were relatively low, their % composition was more of High Molecular Weight that was less readily biodegraded by indigenous microorganisms, and hence can persist in the environment as carcinogens.
Keywords

Biomagnification, Crude Oil Spill, Cassava, Oil Bean, Niger Delta, Carcinogens

1. Introduction

The emergence of oil as the world’s leading fuel was partly due to its relative cleanliness but the enormous scale of the petroleum industry’s operation has inevitably created a new set of difficult environmental problems as being experienced today in the Niger Delta region of Nigeria [1]-[3]. In the study of the socio-economic impact of oil pollution, Worgu [4] stated that crude oil exploration has had adverse environmental effect on soil, forests and water bodies. Oil spill is common fallout of oil exploration and exploitation in the Niger Delta region with an estimated total of over 7000 oil spill accidents reported over a 50-year period [5]. Presently, in Nigeria, oil spills regularly occur in the oil producing areas of the country, while gases are continually flared in these areas [6] [7]. With advanced technology in use in the oil industry, accidents should be less frequent, but this certainly has not completely eliminated accidents and vandalisation [8]. Consequently, the Niger Delta region, over the years, has witnessed massive oil-based environmental degradation and soil fertility loss [9], agricultural decline [10] [11], oil spillage and gas flare, fisheries decline and depletion of biodiversity [12]. Crude oil pollution exerts adverse effects on plants indirectly by making toxic minerals in the soil available to them [13]. Oil spills have also been observed to cause death of plants, and have been linked with blood contamination of people working at impacted sites [14]. Crude oil pollution also leads to deterioration of soil structure, loss of organic matter contents, and loss of soil mineral nutrients. It also exposes soil to leaching and erosion [15]. The enhanced levels of these pollutants are judged in terms of the degree of toxicity, the extent of exploitation of the pollutants, their application, concentration and consequent mobilization into the soil. The presence of these pollutants obviously has resulted to loss of soil fertility, poor crop yield and harmful implications on humans and the entire ecosystem [13]. One of the greatest problems associated with oil pollution is the constant exposure to high concentration of heavy metals from oil [16] and also polycyclic aromatic hydrocarbons (PAHs) in soils from oil polluted areas [17]. Heavy metal contamination of agricultural soils and crops in the vicinity of mining areas has been regarded as a major environmental concern [18]-[20]. The presence of polycyclic aromatic hydrocarbons (PAHs) in food is also a matter of concern that requires continuous monitoring. In recent years, a number of epidemiological studies have shown that a large proportion of human cancers are attributable, at least in part, to dietary factors [21]. Consequently, one of the main reasons for concern about the exposure of humans to environmental contaminants is the evidence that a number of these contaminants are potentially carcinogenic. With the myriads of industries located in the Niger Delta region, it is therefore a thing of interest to investigate the level of heavy metals and PAH distribution in crops grown around these areas. Food crops
grown on contaminated soils take up heavy metals and polycyclic aromatic hydrocarbons and then accumulate them in their tissues. Animals that graze on such contaminated plants and or drink from polluted water bodies accumulate these pollutants in their tissues. Humans that consume heavy metal contaminated foods consequently suffer biochemical disorders [22]. This research aims to evaluate the levels of polycyclic aromatic hydrocarbons and selected heavy metals in crude oil polluted soils and major crops grown on such soils in Ughevuwughe community. This will provide baseline information on the level of risk faced by the people living within that region and for clean-up design.

2. Description of the Study Area

The study was conducted in October 2014 at two different cultivated farmlands from Ekore and Uduvwoku in Ughevuwughe community. Ughevuwughe is an Urhobo community in Uhuire Sub-clan of Ughievwen kingdom (Plate 1) in Ughelli South Local Government Area, LGA Delta State Nigeria. The territory of Ughievwen to which Ughevuwughe belongs is made up of about thirty-two villages before and even after the British came to the area. Ekore and Uduvwoku are oil spill affected clans in Ughevuwughe while Okpe-Olomu, a non-oil impacted clan within same LGA was chosen as control. The immediate neighbours of Ughevuwughe community are Ekrejegba and Ekapamre to the North, Ighwrekan, Edjophe, Otujeremi and Agbowhiame to the South, Ighwrekeka and Usiephron to the West, while to the East, it shares border with Effurun-Otor and Olomu clan. The area lies wholly in the tropics and could be located roughly at 5, 12N and 5, 80E [23]. Ughevuwughe community has no oil well. However, it has Shell Petroleum Development Company (SPDC) facilities, that is, oil pipelines which criss-crosses the village and land. In the year 2010, a major tragedy befell the community. An oil spill occurred on the SPDC Row Utorogu-UPS delivery line, 8km from Utorogu flow station. The delivery line is 16" T/L Utoro-UQCC Magnet-mark MP 4. The spill was a result of ruptured equipment which resulted to leakage of crude oil into streams, farmlands and environment of Ughevuwughe community and other neighbouring communities such as Okpare-Olomu, Otor-Edo, all in the same local government area. The vegetation of the study area falls within the rainforest region of Nigeria. Evidence from the on-the spot assessment of the polluted sites, and inquiries made, shows that no attempt of remediation of the site has been carried out. The disaster affected several kilometres of wet lands and was said to have caused incalculable damage to economic and social lives.

3. Materials and Methods

Materials: Cassava tubers, African oil bean seeds, and soil samples from study locations.

Source: All samples were gotten from Ekore and Uduvwoku (oil spill impacted areas in Ughevuwughe) and Okpe-Olomu (non-oil impacted area), all in Ughelli South Local Government Area, Delta State.
Sample Collection and Preservation

An initial survey was carried out on the site prior to sample collection in order to ensure smooth sampling operations. The samples used in the analysis of this project were cassava tubers (*Manihot esculenta*, Crantz), edible seeds of African oil bean (*Pentaclethra macrophylla*, Benth) and surface soils collected from Udovwoku and Ekore quarters of Ughevweghe, and Okpe-Olomu, all in Ughelli-South Local Government Area, Delta state a South-southern state in Nigeria. A hand soil auger (Nickel-plated carbon steel, 3'' diameter) was used to collect soil samples in random replicates of three, at 0 - 20 cm depth, from two farmlands within the study area, approximately 200 metres apart. Soil samples from the same site were bulked to form a composite sample.
The samples were immediately on collection placed in sterilized, air-tight cellophane bags and labelled. The soil samples were put into appropriate containers and stored in a refrigerator at 4°C prior to laboratory analysis. Soil samples were also obtained in replicates of three from the control site (Okpe-Olomu) with the same geographical terrain as the study area for comparison. The auger was cleaned with water and rinsed with methanol after each sampling.

Samples of cassava tubers and African oil bean were collected randomly from all sites. The choice of the crops was based on their general growth pattern (luxuriant) on the contaminated soil, consumption rate and availability at the study area. For each crop, three replicates of samples were collected. The samples were put into well-labelled sterilized air-tight cellophane bags and stored at 4°C prior until required. Collected crops were identified by a taxonomist in the Department of Plant Biology and Biotechnology, University of Port Harcourt.

4. Preparation of Samples for Analysis

4.1. Preparation of Soil Samples

Soil samples were air-dried under room temperature to ensure constant weight, after which, they were homogenised using a ceramic mortar and pestle to obtain finer texture and to remove sticks, pebbles and rock particles. The air-dried soil samples were then sieved through a 2 mm polythene sieve. Particles larger than 2 mm mesh size were discarded.

4.2. Preparation of Plant Samples

Plant samples of the crops were gently washed under running tap water to remove adhered soil particles, and then with deionized water to remove any possible foliar contaminants such as pesticides, fertilizers, dust or mud. The crops were peeled and then cut into small pieces using a stainless knife, and oven-dried at 60°C for 24 hrs to a constant mass. The dried tissues were stored in a moisture-free environment prior to further processing. The dried samples were then ground using a ceramic mortar and pestle to reduce the dried material to a suitable size for digestion and analysis.

5. Sample Digestion and Analysis

5.1. Extraction and Analysis of Heavy Metals in Soil Samples

The dishes were cleaned, dried and ignited, and covered at 500°C for 30 minutes in the furnace to inactivate or kill all persisting microbes. The dishes were cooled and covered in desiccators. They were weighed until a constant weight was obtained. A 5.0 g of the sieved air-dried soil samples was accurately put into the dish and ignited in a muffle furnace for 6 hours, opening the cover for escape of gases at 500°C. This was checked periodically until complete ashing (a grey-white ash) was obtained. The ash samples were allowed to cool and 5 ml of 10% HCl was added to each sample to enhance dissolution, and 5 ml of 10% HNO₃ was added thereafter and set on a water bath to dissolve.
completely. The solution was evaporated to near dryness on the water bath. On cooling to room temperature, the digest was filtered through Whatman No. 42 filter paper, into a clean, dry 50 ml standard volumetric flask. Both the dish and the filter paper were washed into the flasks, made up to mark with deionized water [24]. The resultant solutions from the respective digestions were ready for metal analysis. The blank was also prepared following the same procedure.

5.2. Extraction and Analysis of Heavy Metals in Cassava Tubers and Oil Bean Seeds

A 5.0 g of each dried and crushed plant sample was accurately weighed (for wet digestion) into dish. This was burnt slightly for 6 hours, opening the cover for escape of gases at 500˚C. It was checked periodically until a grey-white ash was obtained. The ash samples were allowed to cool and 5 ml of 10% HCl was added to each sample to facilitate dissolution, and 5 ml of 10% HNO₃ was added thereafter, and set on a water bath to dissolve completely. The digest was allowed to cool to room temperature and filtered through Whatman No. 42 filter paper, into a clean, dry 50 ml standard volumetric flask. Both the dish and the filter paper were washed into the flasks, and marked up with deionized water [24]. The resultant solutions from the respective digestions and blank, prepared following the same procedure, were used for analysis for Cd, Zn, Ni, Cu and Cr using GBC-Avanta PM SN A6600 Atomic Absorption Spectrophotometer. Matrix matching, standard addition and background correction were used to overcome interference. After every determination, the blanks and certified reference materials were also run to determine the precision and instrumental uncertainty. The percent recoveries of metals from the certified reference material were 93.2%, 95%, 89%, 85% and 96% for Cd, Zn, Ni, Cu and Cr respectively.

5.3. Determination of PAHs

5.3.1. Principle
The samples were extracted with dichloromethane (DCM) and subjected to gas chromatographic analysis (HP Gas Chromatograph 5890 Series II).

5.3.2. Extraction
Two grams of samples were weighed into a clean extraction container (50 ml beaker) and 10 ml of extraction solvent (dichloromethane) was added into the samples, mixed thoroughly, and allowed to settle for 5 minutes. The mixture was carefully filtered into a clean solvent rinsed extraction bottle using filter paper fitted into a Buchner funnel. The extracts were concentrated to 2 ml (using a water bath regulated at 35˚C) and then transferred for clean-up/separation.

5.3.3. Clean Up/Separation
One centimetre of moderately packed glass wool was placed at the bottom of 10 mm ID × 250 Loup chromatographic column. Slurry made from 2 g activated silica in 10 ml methylene chloride was prepared and placed into the chromatographic column. To the top
of the column was added 0.5 centimetre of sodium sulphate. The column was rinsed with additional 10 ml of methylene chloride. The column was pre-eluted with 20 ml of dichloromethane. This was allowed to flow through the column at the rate of about 2 minutes until the liquid in the column was just above the sulphate layer. Immediately, 1ml of the extracted sample was transferred into the column. The extraction bottle was rinsed with 1ml of dichloromethane and added to the column as well. The stop-cork of the column was opened and the eluent/extract was collected with a 10 ml graduated cylinder. Just prior to exposure of the sodium sulphate layer to air, dichloromethane was added to the column in 1 - 2 ml increments. Accurately measured volume of 8 - 10 ml of the eluant was collected and labelled aromatics [25].

5.3.4. Gas Chromatographic Analysis
The concentrated aromatic fractions were transferred into labelled glass vials with Teflon rubber crimp caps for gas chromatographic analysis. One micro litre of the concentrated sample was injected by means of a hypodermic syringe through a rubber septum into the column. Separation occurs as the constituents of the vapour partition between the gas and liquid phases and oven temperature was programmed from 60˚C to 180˚C. The sample was automatically detected as it emerges from the column at a constant flow rate by the FID detector whose response is dependent upon the composition of the vapour.

5.3.5. Statistical Analysis
The results were expressed as mean ± standard error of the mean (SEM). All the data obtained were subjected to one-way ANOVA, using computer aided SPSS package (version 17.0).

5.3.6. Quality Control
Analytically, quality control measures (operation of all instruments in accordance with operating instructions as supplied by manufacturers, unless otherwise specified in the work plan; proper documentation of equipment checkout and calibration activities prior to sampling operations) and recovery study ranging from 92% - 99% were carried out. Safety was generally carefully handled by using appropriate sampling equipment, containers and preservation method to avoid contamination of samples. All glass wares for metal analysis were previously soaked in 14% HNO₃ for 24 hours. All reagents used were of analytical grade and reagent blank determinations were used to correct errors. Multiplicity of samples for each determination ensured reproducibility of data.

6. Results and Discussion
Metals are persistent pollutants that can be biomagnified in the food chain becoming an integral part of life and thus increasingly dangerous to human beings and wildlife. Cadmium, Zinc, Nickel, Chromium and Copper levels were investigated. Cadmium levels in both soils and crop samples from the oil impacted sites were considered elevated (Table 1) in this study. The results showed that Cd levels in soils from Ekore and
Table 1. Mean concentrations of some metals (mg·kg⁻¹) in study samples.

| Contaminants | Amount in soil |                |                |
|--------------|----------------|----------------|----------------|
|              | Okpe           | Ekore          | Uduvwoku       |
| Cd           | 0.001 ± 0.000a | 0.201 ± 0.001c | 0.110 ± 0.000c |
| Ni           | 1.010 ± 0.176a | 2.077 ± 0.003b | 1.263 ± 0.176c |
| Cu           | 0.747 ± 0.007b | 1.050 ± 0.012a | 0.853 ± 0.012a |
| Cr           | 0.003 ± 0.000c | 3.307 ± 0.023a | 4.200 ± 0.025b |
| Zn           | 2.113 ± 0.026a | 0.907 ± 0.009b | 0.477 ± 0.012c |

|                | Amount in Cassava tuber |                |
|----------------|-------------------------|----------------|
| Cd             | BDL                     | 0.240 ± 0.009b | 0.140 ± 0.006b |
| Ni             | 1.963 ± 0.024c          | 2.670 ± 0.210b | 2.230 ± 0.041a |
| Cu             | 1.283 ± 0.009a          | 3.010 ± 0.000b | 3.023 ± 0.009c |
| Cr             | BDL                     | 1.327 ± 0.020b | 3.122 ± 0.688c |
| Zn             | 4.797 ± 0.026a          | 6.433 ± 0.078b | 7.850 ± 0.10b  |

|                | Amount in oil bean seed |                |
|----------------|-------------------------|----------------|
| Cd             | 0.093 ± 0.003a          | 0.137 ± 0.003b | 0.413 ± 0.009b |
| Ni             | 5.593 ± 0.038a          | 2.617 ± 0.018a | 2.480 ± 0.017a |
| Cu             | 0.083 ± 0.024c          | 0.330 ± 0.252b | 3.257 ± 0.013a |
| Cr             | 0.056 ± 0.030b          | 2.543 ± 0.033a | 3.217 ± 0.040a |
| Zn             | 0.243 ± 0.020a          | 1.683 ± 0.030b | 2.250 ± 0.040b |

Values (mean ± S.E.M), n = 3. Means in the same row bearing the same superscript letters are not significantly different at 95% level.

Uduvwoku (oil impacted sites) were significantly higher than Cadmium levels from Okpe (Control site) as shown in Table 1 and exceeded WHO limit of 0.10 mg·kg⁻¹ [26]. Similarly observed Cd levels in Cassava and Oil bean food samples exceeded tolerable weekly intake (TWI) values of 0.003 mg·kg⁻¹ and 0.007 µg·kg⁻¹ set by European Food Safety Authority and Joint FAO/WHO Expert Committee on Food Additives respectively as reported by EFSA [27]. Oil bean samples harvested from control site gave mean value (0.093 mg·kg⁻¹ exceeding reported TWI levels reported by EFSA [27] and may implicate Oil bean meal at bioconcentration of Cd. Cadmium is a non-essential element that negatively affects plant growth and development and is regarded as an extremely significant pollutant due to its high toxicity and large solubility in water as reported by Pinto et al. [28] and Lichtfouse [29]. Observed chlorosis, leaf rolls and stunting in this study were also reported as main symptoms of cadmium toxicity in plants [30].

Cadmium is regarded as a commutative toxin because of the human body’s ability to excrete just 0.001% of the amount ingested in a day and chronic exposures result in kidney damage, bone deformities and cardiovascular problems [31]. Human diseases
have resulted from consumption of Cd contaminated foods [32] as Alloway [33] described its greater mobility and and hence availability to plants in relation to their preferential binding to carbonates. The average biological half-life of Cd, another accumulation poison similar to lead, has been estimated to be about 18 years [34]. Biochemically, Cd has been reported by Wolfgang and Jean-Marc [35] to act as a catalyst in forming reactive oxygen species; increasing lipid peroxidation and additionally depleting antioxidants, glutathione and protein-bound sulfhydryl groups. They also reported promotion of the production of inflammatory cytokines under Cd influence and hence the need for more caution. Overall, the threat heavy metals pose to human and animal health is aggravated by their low environmental mobility, even under higher precipitations, and their long term persistence in the environment [36]-[38]. Observed elevated levels in test areas may suggest that populations feeding on oil bean from these areas may be at a higher risk of Cd toxicity. The accumulation in the system may lead to acute Cd poisoning which includes high blood pressure, kidney damage, destruction of testicular tissue and destruction of red blood cells [39].

The mean concentrations of chromium showed higher accumulations in samples from test sites in relation to control samples with values 0.003 ± 0.000 mg·kg⁻¹, 3.307 ± 0.023 mg·kg⁻¹ and 4.200 ± 0.025 mg·kg⁻¹ for Okpe, Ekore and Uduvwoku respectively. These values fell above the WHO permissible limit of 0.05 mg·kg⁻¹ (WHO, 1985). Soil Cr values were also higher than the critical level of 0.02 mg·kg⁻¹ reported by Anderson et al. (1973). As observed with Cd (Table 1), bioconcentration of available Cr were higher in Oil bean samples even in control samples. Higher levels of Chromium in plants could cause toxicities in human. The permissible limit set by FAO/WHO [40] in edible plants was 0.02 ppm. Chromium has no verified biological role and has been classified as non-essential to mammals [41]. Chromium (VI) is toxic. Acute toxic effects occur when breathing very high levels of chromium (VI) in air, which can damage and irritate the nose, lungs, stomach and intestines [42]. Ingesting very large amounts of chromium can cause stomach upsets and ulcers, convulsions, kidney and liver damage, and even death [43].

Observed Zn levels in study soils were relatively lower than the WHO standard of 100 mg·kg⁻¹ [26]. Zinc concentrations was higher in the food samples from the oil impacted sites. The major sources of zinc in this site is probably the attrition of lubricating oils which zinc is found as part of many additives as zinc dithiophosphates. Zinc is essential to plants and animals in very low concentrations by serving as components of enzymes, structural proteins, pigments and also helping to maintain the ionic balance of cells [44]. Even though zinc is an essential requirement for a healthy body, excess zinc above 25 mg·kg⁻¹ can be harmful, and may cause zinc toxicity.

The observed levels of Copper and Nickel in all soil samples were relatively lower than WHO limits of 10 mg·kg⁻¹ and 10 mg·kg⁻¹ [26] respectively. High doses of copper, above the recommended daily allowance for dietary copper which is 2.5 mg·kg⁻¹ [45], are associated with anaemia, liver and kidney damage as well as irritation of both stomach and intestine [46]. High levels of copper above normal range of 5.00 - 20.00
mg·kg⁻¹ [47] required for plant growth caused grey node symptoms on plants grown on agricultural soils with excess amount of copper.

Nickel was higher in cassava and soil samples from the crude oil contaminated areas except for oil bean samples where that of Okpe (control) accumulated higher. This may be due to accumulation of heavy metals from other sources e.g. the use of fertilizers, waste dumpsites etc. [48].

Crude oil adversely affects the soil ecosystem through adsorption to soil particles, provision of an excess carbon that might be unavailable for microbial use and an induction of a limitation of soil nitrogen and phosphorus [49]. The effects of crude oil pollution on the properties of soil have been the subjects of many studies. Okolo et al. [50] reported that any contact of soil with crude oil results in damage to soil microorganisms and plants while Onuoha et al. [51] among others have shown that crude oil pollution prevents oxygen exchange between the soil and the atmosphere due to hydrophobic properties of oil. Significant delayed and decreased rate of crop germination were reported by Nwaichi and Wegwu [52].

The level of nitrates was 82.500 mg·kg⁻¹ dry weight in control Okpe (control site), 36.200 mg·kg⁻¹ dry weight in Ekore and 14.100 mg·kg⁻¹ dry weight in Uduvwoku for cassava samples (Table 2). This implies that crude oil spill drastically reduced soil nitrate contents of the polluted sites compared to control site at both soil depths. Oil contamination of soil has been shown to limit normal diffusion processes thereby reducing the availability of the level of some nutrients in the soil [53]. Evaluating NPK soil fertility, organic form of Nitrogen Nitrate Nitrogen was moderate for control (Okpe) site and low [54] for test (Ekore and Uduvwoku) sites in this study. However, soil concentrations of NO₃-N depend upon the biological activity and may fluctuate with changes in soil temperature, soil moisture, and other conditions. Considering plant growth, physiology and carbohydrate content, N tops the needed nutrient list. A nitrate nitrogen level of 30 ppm is considered sufficient during the active growing season for most plants [55]. Ammonium nitrogen concentrations fell below 2 - 10 ppm reported as common by Fulton et al. [54] for test sites. They reported that ammonium nitrogen does not usually accumulate in soil because conditions such as soil temperature and moisture suitable for tree growth are also ideal for conversion of NH₄-N to NO₃-N. Higher nitrogen level in the control samples in comparison with the polluted areas could be as a result of fewer disturbances in activities of nitrogen-fixing bacteria and other microorganisms associated with the decomposition of organic matters, which might be inactivated or greatly distressed in the polluted sites. The reduction in the concentration of NO₃-N in the contaminated sites suggests that the process of nitrification might have reduced following the incidence of oil spillage. According to Odu et al. [56], oil degrading or hydrocarbon-utilizing microbes such as Azobacter spp normally become abundant while nitrifying bacteria such as Nitrosomonas spp become reduced in number. This probably explains the relatively lower values of NO₃-N obtained for the contaminated soils. Jobson et al. [57] had earlier reported that oil spills on land resulted in an imbalance in the carbon:nitrogen ratio which, if greater than 17:1 in soils resulted
Table 2. Mean concentrations of nutrient parameters (mg·kg⁻¹) in study samples.

| Nutrients | Amount in soil | Amount in Cassava tuber | Amount in oil bean seed |
|-----------|----------------|-------------------------|-------------------------|
|           | Okpe | Ekore | Uduvwoku | Okpe | Ekore | Uduvwoku | Okpe | Ekore | Uduvwoku |
| NO₃-N     | 20.670 ± 0.880a | 9.880 ± 0.080b | 7.690 ± 0.040b | 82.500 ± 0.981a | 36.200 ± 0.529b | 14.100 ± 0.173b | 145.830 ± 1.975a | 96.433 ± 1.302b | 103.760 ± 1.729b |
| NH₄-N     | 3.274 ± 0.340a  | 1.066 ± 0.066b   | 0.734 ± 0.066b   | 18.636 ± 0.222a  | 8.177 ± 0.017a  | 0.647 ± 0.009b  | 32.943 ± 0.439a  | 21.783 ± 0.292b | 23.440 ± 0.363c  |
| PO₄-P     | 20.000 ± 0.600a | 4.300 ± 0.300b | 2.300 ± 0.300b | 1.077 ± 0.023c  | 0.770 ± 0.017a  | 0.647 ± 0.009b  | 0.737 ± 0.008a  | 0.563 ± 0.008b | 0.263 ± 0.008b  |
| K         | 156.600 ± 7.209a | 99.246 ± 0.046b | 74.440 ± 0.003c | 6455.700 ± 9.241a | 6301.800 ± 6.535b | 3515.700 ± 0.688c | 3387.300 ± 7.458a | 2040.200 ± 9.325b | 1425.900 ± 3.764b |

Values (mean ± S.E.M), n = 3. Means in the same row bearing the same superscript letters are not significantly different at 95% level.

in net mobilization of nutrients by microbes leading to loss of soil fertility. Increase in organic carbon is directly proportional to the increase of crude oil addition to soil. The high C/N ratios leading to immobilization of soil nitrates coupled with the environment brought about by the oil pollution, accounted for low level of NO₃-N in the oil contaminated soil.

Bray P1 method adopted gave marginal fertility in recorded ortho-phosphate level for control and low (<20 ppm) values for test sites. This means that phosphorus availability has been significantly impaired with crude oil spill and poses a growth limiting factor. Also, dry and compacted crude oil spill soils at test sites may have hampered the release of both of Phosphorus. Since soil tests suggested low phosphorus fertility, the possibility of a deficiency was confirmed with plant tissue testing. Soil phosphate levels gave 2.300 ± 0.300 mg·kg⁻¹ in Uduvwoku, 4.300 ± 0.300 mg·kg⁻¹ in Ekore and 20.000 ± 0.600 mg·kg⁻¹ in control site (Table 2). Lower values were also recorded in studied food samples. The reduction in the extractible phosphorus in the crude oil contaminated areas is probably due to high C/P ratio resulting from the crude oil soil addition. The microorganisms which attack the hydrocarbons would immobilize the inorganic phosphorus in the soil, thus bringing about a reduction in extractible phosphorus [58]. This is in agreement with earlier findings of Udo [59].

Extractable Potassium levels in test sites fell within 75 - 150 ppm classified as low by Fulton et al. [54]. Although control site gave moderate amount, it signals future defi-
ciencies ranking in the lower limit of recommended medium range in agricultural soils.

Soil Potassium levels accumulated higher in control samples than in oil contaminated samples with values 74.440 ± 0.003 mg·kg\(^{-1}\) for Uduvwoku, 99.246 ± 0.046 mg·kg\(^{-1}\) for Ekore and 156.600 ± 7.209 mg·kg\(^{-1}\) dry weight for control site (Table 2). For cassava samples, potassium levels gave 3515.700 ± 0.688 mg·kg\(^{-1}\) for Uduvwoku, 6300.800 ± 6.535 mg·kg\(^{-1}\) for Ekore and 6455.700 ± 9.241 mg·kg\(^{-1}\) for control samples. In oil bean samples, potassium levels were 1425.900 ± 3.764 mg·kg\(^{-1}\) for Uduvwoku, 2040.200 ± 9.325 mg·kg\(^{-1}\) for Ekore and 3387.300 ± 7.458 mg·kg\(^{-1}\) in control sample. The decrease in potassium levels of crude oil contaminated soil was similar to the findings of Onuh et al., [60]. It is postulated that the considerable reduction in the availability of K cation in the studied soils are due to physical blockages of the exchange site on the organic and clay mineral components by the hydrocarbon residues [61].

Poorer available P and extractable K could account for poorer tuber yield and chlorotic observations in the farm at Uduvwoku in comparison to Ekore. Similarly, levels of ortho-phosphate and extractable K in Oil bean seeds were higher in samples from Ekore in comparison to Uduvwoku. Plants deficient in potassium are unable to utilize nitrogen and water efficiently and are more susceptible to disease [55]. Higher nitrate and ammonium nitrogen concentrations in Oil bean seeds could be attributed to its membership to the legume family so are able to take large amounts of nitrogen [62] from the air and convert it to protein in the seeds.

The 17 EPA Polycyclic Aromatic Hydrocarbons (PAHs) were investigated in the contaminated soils and control, total mean concentrations of PAHs in mg·kg\(^{-1}\) dry weight of soil, cassava and oil beans and presented in Tables 3-5 respectively. Average individual soil PAHs ranged from BDL to 98.120 ± 0.012 (×10\(^{-3}\) mg·kg\(^{-1}\)) dw, BDL to 643.300 ± 0.116 (×10\(^{-3}\) mg·kg\(^{-1}\)) dw, and BDL to 1380.000 ± 1.732 (×10\(^{-3}\) mg·kg\(^{-1}\)) dw for Okpe, Ekore and Uduvwoku samples respectively (Table 3). Bioaccumulation distribution showed individual PAHs from BDL to 70.640 ± 0.006 (×10\(^{-3}\) mg·kg\(^{-1}\)) dw, BDL to 99.193 ± 0.026 (×10\(^{-3}\) mg·kg\(^{-1}\)) dw and BDL to 614.000 ± 0.058 (×10\(^{-3}\) mg·kg\(^{-1}\)) dw for Okpe, Ekore and Uduvwoku Cassava samples respectively but ranged from BDL to 62.480 ± 0.006 (×10\(^{-3}\) mg·kg\(^{-1}\)) dw, BDL to 106.90 ± 0.05 (×10\(^{-3}\) mg·kg\(^{-1}\)) dw and BDL to 105.300 ± 0.116 (×10\(^{-3}\) mg·kg\(^{-1}\)) dw in the same order (Table 4 and Table 5). Of 17 EPA PAHs assessed, the highest mean concentration was recorded for pyrene in soil and Cassava samples irrespective of location. In oil bean samples however, Benzo (k)fluoranthene ranked highest for Okpe and Ekore samples but Indeno(1,2,3-cd)-pyrene for Uduvwoku samples. Pyrene is ubiquitous in the environment as a product of incomplete combustion of fossil fuels and has been identified in surface and drinking water, numerous foods, and in ambient air [63] [64]. Despite the large body of literature on the toxicity and carcinogenicity of PAHs, toxicity data for pyrene are limited. TRL [65] reported that sub-chronic oral exposure to pyrene produced nephropathy, decreased kidney weights, increased liver weight and slightly haematological changes in mice while White and White [66] documented produced fatty livers in rats. Yoshikawa et al. [67] asserted that a single intraperitoneal injection of pyrene produced swelling.
Table 3. Mean concentrations of PAHs ($\times 10^{-3}$ mg·kg$^{-1}$) in soil samples.

| Contaminants     | Okpe                  | Ekore                  | Uduvwoku               |
|------------------|-----------------------|------------------------|------------------------|
| Acenaphthene     | BDL                   | BDL                    | BDL                    |
| Acenaphthylene   | BDL                   | BDL                    | BDL                    |
| Anthracene       | $69.849 \pm 0.001^a$  | $47.530 \pm 0.011^a$  | $134.30 \pm 0.116^a$  |
| Benzo(a)anthracene | BDL                 | $2.564 \pm 0.001^a$  | $99.120 \pm 0.012^a$  |
| Benzo(a)pyrene   | $0.862 \pm 0.001^a$  | $9.918 \pm 0.001^b$  | $12.330 \pm 0.012^b$  |
| Benzo(g,h,i)perylene | $21.620 \pm 0.012^a$ | $45.160 \pm 0.116^a$ | $281.400 \pm 0.116^a$ |
| Benzo(b)fluoranthene | $5.140 \pm 0.017^a$ | $61.910 \pm 0.017^b$ | $306.500 \pm 0.056^b$ |
| Benzo(k)fluoranthene | $19.360 \pm 0.012^a$ | $100.200 \pm 0.116^a$ | $623.100 \pm 0.058^a$ |
| Chrysene         | $29.380 \pm 0.006^a$ | $19.300 \pm 0.000^a$ | BDL                    |
| Dibenz(a,h)anthracene | $31.700 \pm 0.000^b$ | $17.740 \pm 0.012^a$ | $46.070 \pm 0.012^a$  |
| Fluoranthene     | $62.450 \pm 0.012^a$ | $68.890 \pm 0.000^a$ | BDL                    |
| Indeno(1,2,3-cd)pyrene | $41.300 \pm 0.116^a$ | $59.340 \pm 0.006^a$ | $941.500 \pm 0.058^a$ |
| Fluorene         | $47.310 \pm 0.006^a$ | $34.480 \pm 0.006^a$ | $323.500 \pm 0.000^a$ |
| Naphthalene      | $42.610 \pm 0.006^a$ | $40.800 \pm 0.006^a$ | $204.300 \pm 0.058^a$ |
| 2-methylnaphthalene | $44.450 \pm 0.017^a$ | $18.290 \pm 0.017^a$ | $260.900 \pm 0.231^a$ |
| Phenanthrene     | $17.510 \pm 0.006^a$ | $2.384 \pm 0.012^a$  | BDL                    |
| Pyrene           | $98.120 \pm 0.012^a$ | $643.300 \pm 0.116^a$ | $1380.000 \pm 1.732^a$|
| Total            | $531.661$             | $1261.806$             | $4613.020$             |

Values (mean ± S.E.M), n = 3. Means in the same row bearing the same superscript are not significantly different at 95% level. BDL means below detectable limit.

and congestion of the liver as well as increased serum aspartate amino transferase (AST) and bilirubin levels in rats. U.S. EPA [64] classified Benzo[k]fluoranthene, based on no human data and sufficient data from animal assays, as B2 probable human carcinogen. Benzo[k]fluoranthene produced tumors after lung implantation in mice and when fed with a promoting agent in skin-painting studies and mutagenic in bacteria. U.S. EPA [68] reported that Benzo[k]fluoranthene is a component of mixtures that have been associated with human cancer, despite limited human data specifically linking its exposure to human cancers, and listed these mixtures to include coal tar, soot, coke oven emissions and cigarette smoke.

The total mean PAHs in soil samples from Okpe, Ekore and Uduvwoku were 531.661 ($\times 10^{-3}$ mg·kg$^{-1}$) dw, 1261.806 mg·kg$^{-1}$ dw and 4613.020 mg·kg$^{-1}$ dw (Table 3). Those in cassava samples were 244.698 ($\times 10^{-3}$ mg·kg$^{-1}$) dw, 504.013 ($\times 10^{-3}$ mg·kg$^{-1}$) dw and 1001.464 ($\times 10^{-3}$ mg·kg$^{-1}$) dw from Okpe, Ekore and Uduvwoku respectively (Table 4); while 122.318 ($\times 10^{-3}$ mg·kg$^{-1}$) dw, 446.785 ($\times 10^{-3}$ mg·kg$^{-1}$) dw and 495.815 ($\times 10^{-3}$ mg·kg$^{-1}$) dw were recorded for Oil bean samples from Okpe, Ekore and Uduvwoku respectively (Table 5). Total mean PAHs in soil samples from oil impacted sites (Ekore and Uduvwoku) exceeded Department of Petroleum Resources’ permissible limits of 1000 μg·kg$^{-1}$ [69]. The total mean concentration of PAHs in the crop samples exceeded the European Union Limit of 0.2 mg·kg$^{-1}$ [70]. Similarly, potential danger is inevitable
Table 4. Mean concentrations of PAHs (×10⁻³ mg·kg⁻¹) in cassava samples.

| Contaminants          | Okpe       | Ekore      | Uduvwoku  |
|-----------------------|------------|------------|------------|
| Acenaphthene          | BDL        | BDL        | BDL        |
| Acenaphthylene        | BDL        | BDL        | BDL        |
| Anthracene            | 2.165 ± 0.001a | 4.180 ± 0.001c | 2.682 ± 0.001a |
| Benz(a)anthracene     | BDL        | BDL        | BDL        |
| Benzo(a)pyrene        | 1.000 ± 0.581a | 10.110 ± 0.006c | 10.279 ± 0.000c |
| Benzo(g,h,i)perylene  | 3.530 ± 0.012a | 32.590 ± 0.012c | 22.750 ± 0.017c |
| Benzo(b)fluoranthene  | 2.110 ± 0.006a | 35.500 ± 0.006b | 22.010 ± 0.000b |
| Benzo(k)fluoranthene  | 5.480 ± 0.006a | 42.330 ± 0.017b | 68.770 ± 0.006b |
| Chrysene              | 20.060 ± 0.012a | 15.920 ± 0.012c | 13.980 ± 0.006c |
| Dibenz(a,h)anthracene | 8.683 ± 0.001a | 14.710 ± 0.006a | 14.430 ± 0.012a |
| Fluoranthatne         | BDL        | BDL        | BDL        |
| Indeno(1,2,3-cd)pyrene| 7.800 ± 0.006a | 95.420 ± 0.012a | 43.410 ± 0.006a |
| Fluorene              | 11.330 ± 0.006a | 25.680 ± 0.006a | 35.850 ± 0.017a |
| Naphthalene           | 12.380 ± 0.012a | 61.480 ± 0.012a | 77.990 ± 0.006a |
| 2-methylnaphthalene   | 28.880 ± 0.006a | 56.610 ± 0.006a | 25.983 ± 0.343a |
| Phenanthrene          | 70.640 ± 0.006a | 10.290 ± 0.017a | 28.020 ± 0.000a |
| Pyrene                | 70.640 ± 0.006a | 99.193 ± 0.026b | 614.000 ± 0.058c |
| **Total**             | 244.698    | 504.013    | 1001.464   |

Values (mean ± S.E.M), n = 3. Means in the same row bearing the same superscript are not significantly different at 95% level. BDL means below detectable limit.

As levels of strongly carcinogenic Benzo(a)pyrene marginally exceeded DPR intervention limit of 0.01 mg dw kg⁻¹ for food for Cassava (Uduvwoku and Ekore) and Oil bean (Uduvwoku only) samples. Because of the very low water solubility, PAH will tend to be sorbed to the organic matter in the soil instead of being solubilised in the infiltrating water and through this, it may be transported downwards to the ground water.

IARC [64] identified Fluoranthatne in ambient air, surface, drinking, and waste water, and in char-broiled foods. According to U.S. EPA [68], Fluoranthatne would be expected to be absorbed from the gastrointestinal tract and lungs by analogy to structurally-related PAHs, and can be absorbed through the skin following dermal exposure. They also reported that sub-chronic oral exposure to fluoranthene at doses ≥ 250 mg·kg⁻¹ produced nephropathy, increased liver weights, and increased liver enzyme levels in rats. La Gomes and Liteplo [71] described an in vitro study identified 2-methylfluoranthatne and 3-methylfluoranthatne and their dihydriols as metabolites of fluoranthene. Given limited or no data from animal or human bioassays to assess the carcinogenicity of fluoranthene, U.S. EPA [63] placed fluoranthene in weight-of-evidence group D, not classifiable as to human carcinogenicity.

The very low molecular weight PAHs such as acenaphthalene and acenaphthene were not detected probably due to high volatilization or dissolution which may have
Table 5. Mean concentrations of PAHs (×10−3 mg·kg−1) in oil bean samples.

| Contaminants          | Okpe     | Ekore    | Uduvwoku |
|-----------------------|----------|----------|----------|
| Acenaphthene          | BDL      | BDL      | BDL      |
| Acenaphthylene        | BDL      | BDL      | BDL      |
| Anthracene            | 3.274 ± 0.001a | 4.86 ± 0.001c | 6.907 ± 0.001a |
| Benz(a)anthracene     | BDL      | BDL      | BDL      |
| Benzo(a)pyrene        | 0.894 ± 0.001a | 1.665 ± 0.001a | 12.330 ± 0.012a |
| Benzo(g,h,i)perylene  | 1.620 ± 0.006a | 37.980 ± 0.006a | 35.860 ± 0.012a |
| Benzo(b)fluoranthene  | 1.610 ± 0.006a | 50.120 ± 0.006a | 27.180 ± 0.012a |
| Benzo(k)fluoranthene  | 12.480 ± 0.006a | 106.90 ± 0.05b | 78.300 ± 0.017a |
| Chrysene              | 10.840 ± 0.012a | 26.320 ± 0.012a | 22.030 ± 0.012a |
| Dibenz(a,h)anthracene | 11.060 ± 0.017a | 24.740 ± 0.006a | 28.240 ± 0.012a |
| Fluoranthene          | 15.860 ± 0.012a | 36.980 ± 0.006c | 21.870 ± 0.012a |
| Indeno(1,2,3-cd)pyrene| 5.530 ± 0.006a | 14.540 ± 0.023a | 105.300 ± 0.116a |
| Fluorene              | 11.150 ± 0.006a | 75.130 ± 0.012a | 37.250 ± 0.006a |
| Naphthalene           | 12.140 ± 0.006a | 25.550 ± 0.006a | 42.950 ± 0.006a |
| 2-methylnaphthalene   | 23.260 ± 0.017a | 13.240 ± 0.012a | 29.220 ± 0.012a |
| Phenanthrene          | BDL      | 16.160 ± 0.0129a | 24.238 ± 0.000a |
| Pyrene                | 12.600 ± 0.012a | 12.600 ± 33.333a | 24.140 ± 0.012a |
| Total                 | 122.318  | 446.785  | 495.815  |

Values (mean ± S.E.M), n = 3. Means in the same row bearing the same superscript are not significantly different at 95% level. BDL means below detectable limit.

occurred during the process of extraction. It may also be due to the length time of exposure of the soil samples prior to this study. The results of the concentrations (mg·kg−1) of the PAHs investigated in test soils indicated the presence of benzo[k]-fluoranthene, benzo[b]fluoranthene, pyrene and indeno[1,2,3-cd]pyrene. Benzo[b]-fluoranthene PAH constitutes one of the largest group of compounds with high concentrations in a typical soil sample contaminated with PAHs. From the results, it is evident that the soils of the study sites were contaminated with PAHs at varying concentrations. However, the total PAHs concentrations were very low when compared with the maximum background limits of 15 mg·kg−1 in polluted soils set by Dutch and Polish Environment Ministries respectively [72].

These low values recorded should not be taken for granted; since there is no threshold concentration below which carcinogenic effects of PAH does not occur. Six (6) components of the PAHs (benzo[a]anthracene, chrysene, benzo[b]fluoranthene, indeno[1,3-c,d]pyrene, dibenz[a,h]anthracene and benzo[k]fluoranthene) detected fall within the category of PAHs with the highest risk especially at prolonged exposure [42]. Four (4) out of the six PAHs namely benzo[a]anthracene, chrysene, benzo[b]fluoranthene and benzo[k]fluoranthene are implicated in carcinogenesis according to the Cal-EPA [73]. Consequently, it is feared that the population of people living in and around the study sites (oil spill sites) may be predisposed to high risk of cancer due to
long exposure to PAH compound through contaminated soil and food crops.

The health effects of PAHs have been reviewed extensively. These effects depend mainly on the extent of exposure, dose, innate toxicity and exposure routes [74]. Pre-existing health status and age are other predisposing factors. There is risk of harm in both short and long term exposure. Luch [75] reported associated low IQ (intelligent quotient), childhood asthma, premature delivery, low birth weight, heart malfunction, DNA damages in children linked to cancer at high pre-natal exposure to PAHs.

PAHs are ubiquitous in both urban and rural environment and as a result, it is very common to detect at low soils levels and its toxicity consideration is of utmost importance. Emphasizing of posed threats, Villeneuve et al. [76] showed that heavy PAHs can induce dioxin-like activity and weakened estrogenic responses.

PAHs are classified into two broad groups based on physical and biological properties and include high molecular weight (HMW) and low molecular weight (LMW) PAHs. HMW PAHs consists of 4 - 6 aromatic rings and are less readily biodegraded by indigenous microorganisms, hence can persist in the aqueous environment by bio-accumulating in aquatic organisms like fish and mussels, and are more carcinogenic [77]. Brown and Peake [78] reported that LMW PAHs consists of 2 - 3 aromatic rings and although less carcinogenic also pose toxic effects to many aquatic organisms. Distribution of LMW (Figure 1) and HMW (Figure 2) PAHs in this study revealed mean soil, Cassava and Oil bean LMW PAHs of 221.729, 143.484 and 923 (×10⁻³ mg·kg⁻¹) dw; 125.395, 158.24; and 170.525 (×10⁻³ mg·kg⁻¹) dw and 49.824, 134.94 and 140.565 (×10⁻³ mg·kg⁻¹) dw for Okpe, Ekore and Uduvwoku respectively. In the same vein, HMW PAHs gave mean soil, Cassava and Oil bean levels of 309.932, 1028.322 and 3690.02 (×10⁻³ mg·kg⁻¹) dw; 119.303, 345.773 and 830.939 (×10⁻³ mg·kg⁻¹) dw; and 72.494, 311.845 and 355.25 (×10⁻³ mg·kg⁻¹) dw for Okpe, Ekore and Uduvwoku respectively. Okpe may have been contaminated with naturally occurring PAHs (petrogenic and biogenic origins) hence elevated levels of LMH PAHs. Those for Uduvwoku was significantly higher for soil samples studied (Figure 1). Markedly higher HMW PAHs in Uduvwoku samples is suggestive of pyrolytic PAHs origin and implicate the site over

![Figure 1](image.png)

**Figure 1.** Mean distribution of Low molecular weight (LMW) PAHs in study samples.
Figure 2. Mean distribution of High molecular weight (HMW) PAHs in study samples.

Ekore in terms of more enduring risks implied with less biodegradation of aromatic rings. Of the most cultivated food crops studied, Cassava, one of Nigeria’s most staple food is highly implicated in bioaccumulation of recalcitrant and more carcinogenic HMW PAHs.

7. Conclusion

From the results of this study, it is evident that oil exploration and production activities (and oil spillages) variably degraded the soil where such activities are carried out. The study has demonstrated that crude oil contamination can lead to gradual heavy metal and PAHs build up in plants growing in such soils as up to 0.413 mg·kg⁻¹ of Cadmium is found in Oilbean harvested from impacted site. Besides recorded nutrient loss, real effect of spill on soil in terms of soil fertility may have been cushioned by agricultural activities and could be worse on inactive soils. The plants studied were highly vulnerable, suggesting that the site poses potential hazards to grazing animals, humans and the food chain. There is, therefore the urgent need for remediation processes or strategies and management of the contaminated soils, so as to render it fit for use especially for agricultural and domestic purposes. Data generated from study soil testing could inform degree of contamination, efficient and effective resource management of Ekore and Uduvwoku farmlands.

Acknowledgements

The authors will like to acknowledge Okonkwo, I. C. for assistance in data acquisition.

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