Characterization of the State of Hydrogen at High Temperature and Density

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Abstract

Fermionic path integral Monte Carlo simulations have been applied to study the equilibrium properties of the hydrogen and deuterium in the density and temperature range of $1.6 < r_s < 14.0$ and $5000K < T < 167000K$. We use this technique to determine the phase diagram by identifying the plasma, the molecular, atomic and metallic regime. We explain how one can identify the phases in the path integral formalism and discuss the state of hydrogen for 5 points in the temperature-density plane. Further we will provide arguments for the nature of the transitions between the regimes.

1 Introduction

The phase diagram of hydrogen has been studied intensively with different theoretical approaches [1],[2], simulation techniques [3],[4] and experiments [5],[6]. From theory, the principal effects at low densities are well-known. On the other hand, the properties at intermediate density are not yet well understood, and the phase diagram is not yet accurately determined. In particular, the nature of the transition to a metallic state is still an open question.

In this article, we would like to show how these questions can be addressed by path integral Monte Carlo (PIMC) simulations. Using this approach, we derived the phase diagram in Fig.1 where we distinguish between molecular, atomic, metallic and plasma regimes. We will demonstrate how these different states can be identified from PIMC simulations.

The imaginary-time path integral formalism [7] is based on the position-space density matrix $\rho(\mathcal{R},\beta)$, which can be used to determine the equilibrium expectation value of any operator $\hat{O}$,

$$\langle \hat{O} \rangle = \frac{\text{Tr} \hat{O} \rho}{\text{Tr} \rho} = \frac{\int d\mathcal{R}d\mathcal{R}' \rho(\mathcal{R},\mathcal{R}',\beta) \langle \mathcal{R}|\hat{O}|\mathcal{R}' \rangle}{\int d\mathcal{R} \rho(\mathcal{R},\beta)}$$

where $\mathcal{R}$ represents the coordinates of all particles. The low temperature density matrix $\rho(\mathcal{R},\mathcal{R}',\beta) = \langle \mathcal{R}|e^{-\beta \mathcal{H}}|\mathcal{R}' \rangle$ can be expressed as product of high temperature density matrices $\rho(\mathcal{R},\mathcal{R},\tau)$ with the time step $\tau = \beta/M$. In position space, this is a convolution,

$$\rho(\mathcal{R}_0,\mathcal{R}_M;\beta) = \int \cdots \int d\mathcal{R}_1 d\mathcal{R}_2 \cdots d\mathcal{R}_{M-1} \rho(\mathcal{R}_0,\mathcal{R}_1;\tau) \rho(\mathcal{R}_1,\mathcal{R}_2;\tau) \cdots \rho(\mathcal{R}_{M-1},\mathcal{R}_M;\tau).$$

This high dimensional integral can be integrated using Monte Carlo methods. Each particle is represented by a closed path in imaginary time. Fermi statistics is taken into account by considering the fermion density matrix, which can be expressed by considering all permuta-
Figure 1: The computed phase diagram of deuterium is shown in the temperature-density plane. (×, ⋄, △) indicate our PIMC simulations and distinguish between different degrees of degeneracy of the electrons (× less than 10% exchanges, ⋄ more 10% and △ over 80%). The four main regimes, molecular, atomic and metallic fluid as well as the plasma are shown. The thick solid line specifies the plasma phase transition predicted in [9]. The thin solid lines specify the approximate location of isobars.

The phase diagram is generated using PIMC simulation with 32 protons and 32 electrons and a time step $\tau = 1/10^6$ K. In the low density and low temperature regime, we find a molecular fluid. In the proton-proton correlation function shown in Fig. 2, one finds a clear peak at the bond length of 0.75 Å. We determine the number of molecules as well as other compound particles by a cluster analysis based on the distances. Using this approach we can estimate the number of bound states (see [10]). We can also estimate...
Figure 2: Distribution functions for a selection of 5 simulation of hydrogen at different temperatures and densities, one in each column: 1) a plasma, 2) a atomic fluid, 3) a molecular fluid, 4) a molecular fluid with metallic properties, and 5) a metallic fluid. The rows show the following: (1) proton-proton correlation function $g_{pp}(r)$ multiplied by the density, which means the area under the peak at the bond length of $r = 0.75\text{Å}$ indicates the number of molecules, (2) $[g_{pe}(r) - 1]r^2$, where the first peak hints to the existence of bound electrons in the ground state, (3) pair correlation function for electrons with parallel spins demonstrating the Pauli exclusion principle, (4) pair correlation function for electrons with anti-parallel spins, where the peak is caused by a localization of wave function along the molecular bond, and (5) distribution of the fraction $\nu$ of electrons involved in a permutation. A peak near $\nu = 0$ represent a small degree of degeneracy of the electrons, while one near $\nu = 1$ implies a highly degenerate electron gas.
the fraction of molecules and atoms to determine the regime boundaries. However at high density, a clear definition of those species is difficult to give.

Starting in the molecular regime, one finds that increasing temperature at constant density leads to gradual dissociation of molecules followed by a regime, with a majority of atoms. The atoms are then gradually ionized at even higher temperatures. Lowering the density at constant temperature leads to a decrease in the number of molecules, or atoms respectively, due to entropy effects.

If the density is increased at constant temperature, pressure dissociation diminishes the molecular fraction. This transition was described by Magro et. al. [9]. Its precise nature is still a topic of our current research. Using PIMC simulations, one finds it occurs within a small density interval and we predict that it is connected with both the molecular-atomic and insulator-metal transition. We determine the fraction of electrons involved in a permutation as an indication of electronic delocalization. Permuting electron are required to form a Fermi surface, which means that a high number of permutations indicate a high degree of degeneracy of the electrons. Permuting electrons form long chains of paths and therefore occupy delocalized states. This delocalization destabilizes the hydrogen molecules. Before all bonds are broken, one finds a molecular fluid with some permuting electrons, which could indicate the existence of a molecular fluid with metallic properties.

The boundaries of the metallic regime are determined by two effects. With increasing temperature, the degree of degeneracy of the electrons is simply reduced. If the temperature is lowered, the attraction to the protons becomes more relevant, which localizes the electron wave function and decreases the degree of degeneracy also (see Fig. [1]).

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