Evaluation of the Bitterness-Masking Effect of Powdered Roasted Soybeans

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Abstract: The masking of bitterness is considered important because many pharmaceutical compounds have a bitter taste. The bitterness-masking effect of powdered roasted soybeans (PRS) was investigated using a bitter taste sensor. PRS was revealed to significantly suppress the bitterness of quinine hydrochloride and denatonium benzoate. Furthermore, the bitterness-masking mechanism of PRS extracts was evaluated using dynamic light scattering. These results showed that the extracted suspension consisted of particles that were several hundreds of nanometers in size. Analysis of the PRS extracts by nuclear magnetic resonance spectroscopy indicated that denatonium benzoate was entrapped in the PRS extracts. Thus, PRS may be useful as a bitterness-masking agent in orally administered pharmaceuticals.

Keywords: powdered roasted soybean; bitterness; masking effect; quinine hydrochloride; denatonium benzoate

1. Introduction

The five basic tastes comprise sweet, salty, sour, bitter, and umami. In particular, sour and bitter tastes are generally unfavorable and avoided by humans because toxic substances often taste bitter [1,2]. Bitter-tasting foods are not preferred in most cases, with a few exceptions including coffee, beer, and wine. Even though humans are averse to bitter-tasting substances, pharmaceutical compounds with physiological benefits often taste bitter. Therefore, the masking of bitterness is considered important in food processing and pharmacology.

Several bitterness-masking methods have been developed, such as the addition of other tastes and flavors to suppress bitter tastes [3–5]. Bitterness can also be masked by applying antagonists for bitter taste receptors (T2Rs), which are classified as G protein–coupled receptors [6–9]. Moreover, coating and encapsulating are often used in the pharmaceutical industry to mask the bitterness of drugs [10–17]. The formation of inclusion complexes between cyclodextrin and various substances can be used to mask bitterness [18–20]. Furthermore, phosphatidic acid and its lipoprotein derivative formed by interactions with β-lactoglobulin are reported to suppress the bitterness of quinine [21]. In addition, amino acid derivatives are low-molecular-weight bitterness-masking compounds [22]. In many cases, however, the bitterness-masking mechanisms of these compounds have not been elucidated. For food processing, these compounds must be harmless; therefore, identifying safe bitterness-masking agents originating from foods is desirable.
Powdered roasted soybeans (PRS), called kinako in Japan, are a traditional Japanese food. Each 100 g of PRS is composed of 39 g of protein, 25 g of total lipid, 30 g of carbohydrate, and small amounts of minerals and vitamins [23]. PRS is inexpensive and easily handled in its powder form. Therefore, we thought that if a bitterness-masking effect could be confirmed, PRS would be a useful bitterness-masking agent. In a previous study, we used a sensory evaluation test to investigate the bitterness-masking effect of PRS [24]. This test revealed that PRS masks many components of bitterness, and it was particularly effective in masking the bitterness of hydrophobic components, such as quinine hydrochloride (QH) and denatonium benzoate (DB). However, other powdered foods such as starch did not mask the bitterness in this test.

In this study a bitter taste sensor (Intelligent Sensor Technology, Inc., Kanagawa, Japan), a powerful tool for quantifying bitterness-masking [25–31], was used to confirm the bitterness-masking effect of PRS previously found in sensory tests. Furthermore, the bitterness-masking mechanism was evaluated by extracting the masking components from PRS and assessing them using dynamic light scattering (DLS) and nuclear magnetic resonance (NMR) analyses.

2. Materials and Methods

2.1. Reagents

Powdered roasted soybeans were obtained from Kawamitsu Bussan Co., Ltd. (Tokyo, Japan), quinine hydrochloride was obtained from Nacalai Tesque Inc. (Kyoto, Japan), and denatonium benzoate was obtained from Tokyo Chemical Industry Co., Ltd. (Tokyo, Japan).

2.2. Bitter Taste Sensing

A taste sensing system (TS-5000Z, Intelligent Sensor Technology, Inc., Kanagawa, Japan) was used to measure bitter ingredients [25–27]. The sensor has a working electrode with a lipid/polymer membrane for sensing and a reference electrode. Changes in the membrane potential generated when the working electrode is immersed in each sample are measured. The bitterness of 0.2 mg/mL QH solution and 0.02 mg/mL DB solution were measured by the change in the membrane electric potential when the bitter ingredients were absorbed into the membrane. To measure the reference potential (Vr), the sensor electrode was immersed in a 30 mM KCl solution. Subsequently, the sensor electrode is immersed in the sample solution to measure the membrane potential (Vs). The difference between these potentials, Vs – Vr, is defined as the sensor output. Solutions containing masking ingredients PRS (10 or 15 mg/mL) and OH (0.2 mg/mL) or DB (0.02 mg/mL) were prepared, and the sensor output was obtained for each sample.

2.3. Dynamic Light Scattering

PRS (10.0 g) was added into 100 mL of chloroform. The heterogeneous solution was stirred for 1 h, and then filtered. The organic layer was evaporated and 1.0 g of residue was obtained. This residue was purified using gel filtration chromatography on Sepharose 4B (GE Healthcare Bio-Science KK, Tokyo, Japan). The PRS-extracts were dissolved in phosphate buffered saline solution (pH 7.4, 0.1 mM) and evaluated by DLS. The diameters of the particles were measured using a Zetasizer Nano ZS ZEN3600 (Malvern Instruments Ltd., Worcestershire, UK).

2.4. Nuclear Magnetic Resonance

DB (5.0 mg), extracted PRS (10.0 mg), or DB (5.0 mg) and extracted PRS (10.0 mg) were dissolved in 0.6 mL of deuterated water (D2O). The 1H NMR spectrum of each sample was recorded at 400 MHz with a JEOL-400 spectrometer (JEOL Ltd., Tokyo, Japan).
2.5. Statistical Analysis

The bitter taste sensing data were statistically analyzed using the Student’s $t$-test. Differences from the control conditions were considered significant at $p$-values of less than 0.05* and 0.01**.

3. Results

The bitterness of various samples was evaluated using a bitter taste sensor (Figure 1). The sensor output for a 0.2 mg/mL QH solution was 84.9 mV. When 10 and 15 mg/mL of PRS were added to QH, the sensor output for the samples decreased to 54.8 and 48.6 mV, respectively. Similarly, the sensor output of 30.6 mV for a 0.02 mg/mL DB solution decreased to 26.6 and 24.4 mV when 10 and 15 mg/mL PRS, respectively, were added to DB.

![Figure 1](image1.png)

**Figure 1.** The bitterness of QH and DB in the absence and presence of powdered roasted soybean (PRS) measured as the change in the membrane electric potential of a taste sensor. The reference potential was obtained by immersing the sensor electrode in a 30 mM KCl solution. The bars represent mean ± SD, $n = 5$, * $p < 0.05$, ** $p < 0.01$.

The extracts of PRS were analyzed using DLS, as shown in Figure 2. The solution of substances extracted from PRS in phosphate buffered saline solution (pH 7.4, 0.1 mM) was found to contain particles of over 100 nm in size.

![Figure 2](image2.png)

**Figure 2.** DLS analysis of 1.69 mM PRS extracts in phosphate buffered saline solution (pH 7.4, 0.1 mM).

The $^1$H NMR spectra of DB, the substances extracted from PRS, and a mixture of DB and the extracted substances were measured in aqueous solution, as shown in Figure 3. The DB peaks in the mixture were broadened when compared to those of the DB solution.
4. Discussion

Quantitative evaluations of the degree of bitterness using a bitterness sensor showed that PRS significantly suppresses the bitterness of QH and DB. This result indicates that the bitterness taste receptor does not participate in the QH or DB suppression mechanism of PRS. Moreover, the quantitative result was consistent with the previous sensory study [24] on the masking effect of PRS on QH and DB. On the other hand, in the previous sensory study, other powdered foods such as starch did not mask the bitterness, indicating that the bitterness suppression mechanism does not involve physically covering the tongue with powdered solids. Therefore, to further elucidate the bitterness-masking mechanism of PRS, we focused on the ability of PRS to form a suspension in water. The substances in the PRS suspension were extracted using chloroform. The PRS extracts also suppressed the bitterness of QH and DB to a similar extent as that observed for PRS using the bitterness sensor. The PRS extracts were evaluated using DLS (Figure 2), and the suspension was found to consist of particles that were several hundreds of nanometers in size. To evaluate these particles in the suspension of PRS extracts, $^1$H NMR spectra were collected (Figure 3). The observation of several peaks between 0.8 and 2.6 ppm indicated the presence of hydrocarbons derived from fatty acids in PRS extracts. As the suspension is in aqueous solution, these fatty acids are likely derived from phospholipids, which are abundant in PRS. Moreover, the broadening of all the peaks for the PRS extracts indicates that polymerized structures were formed. These NMR and DLS results suggest the formation of liposomes that are several hundreds of nanometers in size by the association of phospholipids [32]. Interestingly, the addition of DB into the suspension of PRS extracts resulted in the broadening of the sharp peaks of DB (1.4, 3.5, 4.1, 4.8, and 7.1–7.8 ppm), similar to the broad PRS extract peaks (Figure 3b). This result indicates that DB interacts with PRS extracts via hydrophobic and ionic interactions [33]. QH is expected to interact with PRS through the same mechanism. Therefore, we think that the bitterness-masking mechanism of PRS might involve the bitterness components becoming entrapped in liposomes formed by the phospholipids of PRS [34,35]. Thus, the bitterness components are physically removed from T2Rs.

In this study, a bitter taste sensor analysis was used to reveal the bitterness-masking effect of PRS. The bitterness-masking mechanism was evaluated by analyzing the masking components extracted from PRS using DLS and NMR. PRS has many components in addition to phospholipids, which may contribute to further masking mechanisms that are not yet understood. However, as the phospholipids

![Figure 3](image-url). The $^1$H NMR spectra (400 MHz, 298 K, D$_2$O). (A) DB in D$_2$O solution; (B) PRS extracts + DB in D$_2$O solution; and (C) PRS extracts in D$_2$O solution.
of PRS are easily handled and undergo hydrolysis due to lipase present in the duodenum, PRS might be useful as a component of orally administered medicines that are absorbed after leaving the duodenum.

Author Contributions: Y.M. supervised the work and wrote the manuscript. T.I., N.K., M.F., K.F., and R.M. carried out the experiments and evaluated the data. J.F. and S.F. were involved in discussions of the experimental results and critical reading of the manuscript.

Conflicts of Interest: The authors declare no conflicts of interest.

Abbreviations
The following abbreviations are used in this manuscript:

| Abbreviation | Description                               |
|--------------|-------------------------------------------|
| PRS          | Powdered roasted soybeans                |
| T2Rs         | Bitter taste receptors                   |
| QH           | Quinine hydrochloride                    |
| DB           | Denatonium benzoate                      |
| DLS          | Dynamic light scattering                 |
| NMR          | Nuclear magnetic resonance               |
| Vr           | Reference potential                      |
| Vs           | Membrane potential                       |

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