Taxis Toward Hydrogen Gas by

*Methanococcus maripaludis*

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Knowledge of taxis (directed swimming) in the *Archaea* is currently expanding through identification of novel receptors, effectors, and proteins involved in signal transduction to the flagellar motor. Although the ability for biological cells to sense and swim toward hydrogen gas has been hypothesized for many years, this capacity has yet to be observed and demonstrated. Here we show that the average swimming velocity increases in the direction of a source of hydrogen gas for the methanogen, *Methanococcus maripaludis* using a capillary assay with anoxic gas-phase control and time-lapse microscopy. The results indicate that a methanogen couples motility to hydrogen concentration sensing and is the first direct observation of hydrogenotaxis in any domain of life. Hydrogenotaxis represents a strategy that would impart a competitive advantage to motile microorganisms that compete for hydrogen gas and would impact the C, S and N cycles.

Hydrogen gas (H2) is a crucial substrate for methanogens as well as a common source of energy for other organisms in both anaerobic and aerobic environments, including acetogens, sulfate- and sulfur-reducers, and hydrogen-oxidizers1–4. Biological methane (CH4) production from H2 and carbon dioxide (CO2) contributes to greenhouse gas emissions and is possibly one of the oldest microbial metabolisms5,6. Understanding the ecological strategies of methanogens is not only important for our knowledge of early earth processes and present-day anaerobic environments, but also for determining potential roles in human health conditions (e.g., colon cancer and periodontal disease), where positive correlations have been made with incidence of disease and occurrence of methanogens7,8. *Methanococcus maripaludis* is an anaerobic archaea that can use H2 or formate as electron donor to reduce CO2 to CH4 and is considered a model mesophilic methanogen. Recently, the swimming behavior of *M. maripaludis* was described9, but chemotactic responses have not been shown. Chemotaxis has been demonstrated for Archaea, including methanogens10,11, but taxis to hydrogen has not been shown for any domain of life. The chemotaxis signal transduction system in *Archaea* is similar to the well-studied system in *Bacteria*; however, the flagellar switch is different and none of the archaeal flagellar proteins have homologs to bacterial flagellar proteins12–15. Chemotaxis has been the subject of many mathematical models and the majority have concentrated on reproducing the population-level observation of migrating bands of high cell concentration in swarm plates and capillary experiments16. Pioneering work in modeling chemotaxis behavior by Keller and Segel in 1971 has been the basis of the most common mathematical models17. In one dimension, with x being the spatial variable, the Keller-Segel model can be described as a flux, J, such that

\[ J = -\mu \frac{\partial b}{\partial x} + \chi(s) \frac{\partial s}{\partial x} \]  

where \( \mu \) is the cell diffusion coefficient that takes random, non-directed, movement of cells into account. \( b \) is the microbial population density, \( s \) is the attractant concentration and \( \chi(s) \) is the non-constant chemotactic coefficient. The population flux, \( J \), can be differentiated to yield the more common form

\[ \frac{\partial b}{\partial t} = -\frac{\partial J}{\partial x} = -\frac{\partial}{\partial x} \left( -\mu \frac{\partial b}{\partial x} + \chi(s) \frac{\partial s}{\partial x} \right) \]  

and the average cell swimming velocity, \( v \), is calculated by dividing the flux by the population density, or \( v = J/b \). Lapidus and Schiller18 proposed a form of \( \chi(s) \) such that...
where $\chi$ is the constant chemotactic coefficient and $k_d$ is the receptor-ligand binding dissociation constant. The Lapidus-Schiller $\chi(s)$ term, and variations thereof, have been used widely to describe chemotaxis in bacteria. Most work on archael chemotaxis has been performed with Halobacterium salinarum; however, attempts to mathematically describe population flux in archaea have not been published. In addition, $\chi$ and $k_d$ have not been determined for any archaea.

The goal of the present study was to subject M. maripaludis cells to a H$_2$ concentration gradient and compare swimming behavior to model predictions. Attractant (H$_2$) transport was modeled by Fickian diffusion and consumption by the population was modeled in the Michaelis-Menten form such that

$$\frac{\partial s}{\partial t} = D \frac{\partial^2 s}{\partial x^2} - \frac{r_{\text{max}} bs}{(k_m + s)}$$

where $D$ is the diffusion coefficient for H$_2$, $r_{\text{max}}$ is the maximum consumption rate, and $k_m$ is the half-saturation constant.

A modified capillary assay was used in which cells were loaded into a gas-tight anaerobic capillary under an anoxic atmosphere, and a valve allowed controlled exposure to a H$_2$ source (see Supplementary Fig. S1 and S2 online). In previous experiments, capillaries that contained a dissolved chemoattractant were immerged into a cell suspension and cells entered the capillary in the presence of a chemoattractant over a known incubation time. The population increase in the capillary relative to a control was then quantified through cell enumeration. For this study, a novel method was developed to directly observe microscopic swimming behavior of anaerobic cells inside a capillary during exposure to a gas. M. maripaludis cells were tracked for direction and velocity changes upon exposure to H$_2$ or an Argon (Ar) control to quantify the average population-wide response. Cell movement was measured using time-lapse confocal laser scanning microscopy (CLSM) in the center of the capillary (0.5 cm from the gas phase; as in Supplementary Fig. S1 online). A reaction-diffusion model predicted that within 10–15 minutes H$_2$ concentrations would reach the threshold of 2.5–23 $\mu$M, at which hydrogenotrophic methanogens have been shown to use H$_2$ in pure culture studies.

**Results**

When M. maripaludis cells were exposed to H$_2$, the average swimming velocity exhibited significant bias toward H$_2$ within ten minutes (Figure 2A). Biased random walk was not observed when cells were exposed to an Ar gradient (Figure 2B), nor was swimming velocity affected normal to the H$_2$ or Ar gradients (Figures 2C and D). A strong chemotactic response was observed when cells were first starved of H$_2$ for 4–5 hours, while cultures that had not been starved did not show an increase in biased swimming (see Supplementary Fig. S3 online). A reaction-diffusion model predicted that within 10–15 minutes H$_2$ concentrations would reach the threshold of 2.5–23 $\mu$M, at which hydrogenotrophic methanogens have been shown to use H$_2$ in pure culture studies.

![Figure 1](image1.png) **Figure 1** The predicted hydrogen concentration over time at the observation point 0.5 cm from the gas phase over the course of the experiment shown with (A) linear axes and (B) log-linear axes. The 2.5–23 $\mu$M threshold, at which hydrogenotrophic methanogens have been shown to use H$_2$ in pure culture studies, is reached at approximately 10 minutes.

A Keller-Segel model was applied with boundary conditions specific to our study in an effort to quantify the observed chemotactic response to H$_2$ in context with other known chemotactic responses. Currently, K-S model parameter values do not exist for any organism in Archaea or for any other gas besides O$_2$; therefore, a broad range of $k_d$ values were used in the model. Three unknowns, namely ligand-receptor dissociation constant ($k_d$), chemotactic coefficient ($\chi$), and random cell diffusion coefficient ($\mu$), were independently fitted by varying one unknown across the range of published literature values, while keeping the other two variables constant at the average literature values (see Supplementary Table S1 online). The best fit for average swimming velocity was obtained with a $k_d$ value of 0.70 mM, with 0.30 and 2.30 mM corresponding to the 95% confidence interval of the data, while $\chi$ and $\mu$ were kept at average published values (Figure 4A). This range for $k_d$ is similar to that observed for Escherichia coli AW405 to $\alpha$-methyl aspartate. It is, however, quite different from values reported for Bacillus subtilis receptor affinity to O$_2$ ($0.0015$ and $0.075$ mM for the high and low affinity of the receptor, respectively), which is the only previously reported $k_d$ for a gas.

The model could also be fitted to the velocity curve by varying $\chi$ and keeping $k_d$ and $\mu$ constant at average literature values (Figure 4B). The average $\chi$ of $9.3 \times 10^{-3}$ cm$^2$ s$^{-1}$ used to fit the experimental results is higher than the literature range $7.20 \times 10^{-3}$ to $1.24 \times 10^{-3}$ cm$^2$ s$^{-1}$ (see Supplementary Table S1 online). The

$$\chi(s) = \frac{k_d}{(k_d + s)^2}$$

was higher for non-starved cells ($9.3 \mu$m$^{-1}$ versus $8.5 \mu$m$^{-1}$ for starved cells) and lowest for starved cells after exposure to H$_2$ (Figure 3B). The highest maximum observed speed was 91 $\mu$m s$^{-1}$ for H$_2$ starved cells after exposure to H$_2$ (Figure 3B inner boxes), approximately twice the previously observed maximum speed of 45 $\mu$m s$^{-1}$ for M. maripaludis.

The average swimming speed of starved cells was $2.1 \pm 0.05$ $\mu$m s$^{-1}$ before exposure to H$_2$ (Figure 3A) and increased to $3.1 \pm 0.02$ $\mu$m s$^{-1}$ after exposure; equal to non-starved cells. Maximum swimming speed averages (calculated by averaging the single maximum speed from each time point) were highly variable between time points, and the average maximum swimming velocity exhibited significant bias toward H$_2$ within ten minutes (Figure 2A). 3140 | DOI: 10.1038/srep03140
experimental results were best fitted to the model that assumed the average published value for $x$, and then by varying $k_d$ (Figure 4A). The model could not be fitted to the experimental data by varying $m$ according to published values (Figure 4C, note the difference in y-axis scale); however, changing $m$ allowed for a change in shape of the velocity curve.

The response time predicted by the model does not fit the rapid response observed in the experiment and this is likely due to an inaccurate prediction of mass transport inside the capillary (for both $H_2$ and cells). Both random cell diffusion ($\mu$) and the diffusion of hydrogen would require correction from predicted values if mass transport was inaccurately represented. First, considering $\mu$ independently, it is likely that response time would be faster for cells with a larger $\mu$ (Figure 4C). This is based on the proportionality of $\mu$ to swimming speed and run time, and inverse proportionality to one minus the cosine of the turn angle $\theta^{31}$. Although not measured directly here, M. maripaludis has been shown to have relatively long runs and small changes in direction or turn angles$^{9}$. Turn angles between 20 and 45° would result in the largest value of $\mu$ for a given swimming speed and run time, so it is reasonable to assume $\mu$ would be higher for this type of swimming. The observed rapid response can be explained by an inaccurate prediction of $H_2$ mass flux into the liquid domain. If $H_2$ were to reach the observation point faster than predicted by the model, then one would predict a proportionally faster response. This was an entirely static system on a visually observable scale but it is possible that unpredictable micro-scale convective forces increased $H_2$ mass transport. The collective, directed motion of swimming cells may have induced convective flow inside the capillary. Previous work estimates that a force of 0.5 pN is exerted by a cell swimming at 25 μm s$^{-1}$ $^{32}$. While

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**Figure 2** | Average cell velocity (black) with 95% confidence intervals (gray) before and after opening valve to gas phase for (A) $H_2$ and (B) Ar control. Positive y-axis values indicate movement toward the gas phase, negative y-axis values indicate movement away from the gas phase. Average cell velocity (C) and (D) normal to concentration gradient ($n = 5$ for $H_2$ and $n = 3$ for Ar).

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**Figure 3** | Swimming speed of M. maripaludis cells when starved for 5 hours before $H_2$ exposure (−$H_2$ and $n = 4,725$), after $H_2$ exposure (+$H_2$ and $n = 20,189$), and when not starved (No starve, $n = 45,001$). Values represent the mean and error bars represent 95% confidence intervals. (A) Average swimming speed. The difference between −$H_2$ and both +$H_2$ and no starvation is significant ($p < 0.05$). (B) Average maximum swimming speeds are represented by bars, and absolute maximum values inside each bar represent the highest observed speed over all time points for the condition. Differences between all conditions are significant ($p < 0.05$) and $p$ values were calculated with two-tailed t-tests.
chemotaxis-induced convective flow is difficult to demonstrate conclusively, increased mass transport in the capillary could affect mass transport predictions.

A diffusion coefficient correction term, $d$, was introduced as a tool to investigate the effect of increased mass transport on the predicted response time. The correction was applied to the H$_2$ diffusion coefficient and the random cell diffusion coefficient ($\mu$) simultaneously, making the assumption that micro-scale convection was the primary source of error. Any convection would affect the random diffusion of cells and the diffusion of H$_2$ equally. Increased mass transport in the liquid (simulating convection) caused an earlier response in all conditions that were investigated (see Supplementary Fig. S4 online). The value of $d$ is likely to be on the order of $1 \times 10^{-9}$ to $1 \times 10^{-8}$ cm$^2$ s$^{-1}$ based on curve shape and response time as compared to the experimental velocity data.

Another possible explanation for the more rapid experimental response is the presence of more than one type of H$_2$ receptor with varying affinities. This would allow *M. maripaludis* to respond to H$_2$ across a wider range of concentrations. The predicted $k_d$ value 0.7 mM is high, and the presence of a high affinity receptor in *M. maripaludis* is likely. *B. subtilis* O$_2$ receptors with two distinct affinities and binding components have been shown$^{30}$. The presented data demonstrate a chemotactic response to H$_2$, but more focused physiological studies and subsequent model refinement are needed to better represent this poorly understood phenomenon.

In our capillary assay, cells were not observed to accumulate in bands despite the chemotactic response observed. Typical chemotactic bands observed in capillary and swarm plate assays result from cell accumulation as a net result of a biased random walk$^{16}$. In the capillary assay used, convection inside the capillary may have prevented cell accumulation. It is also possible that there are factors involved in chemotactic band formation that are unique between organisms or factors specific to hydrogenotaxis (e.g., swimming mode, adaption response, quorum sensing).

In our work, banding-like behavior was observed on a larger scale when *M. maripaludis* batch cultures were grown statically with H$_2$. Under these conditions a pellicle formed at the gas-liquid interface (Figure 5A, B). This is a similar observation to the one made by Beijernck in 1893, where aerotactic cells were observed swimming toward the meniscus of a test tube$^{16}$. When cultures of *M. maripaludis* were grown statically with the soluble electron donor formate, cells grew throughout the liquid medium (not in a pellicle) and had less cell-associated carbohydrate than H$_2$-grown pellicle cultures (see Supplementary Fig. S5 online). These results suggest that
extra-polymeric substance (EPS) production is important for the formation of a pellicle in response to a H₂ gradient and may require a time-scale longer than used in the capillary experiment.

**Discussion**

The hydrogenotrophic methanogen, *M. maripaludis*, displays chemotactic behavior toward H₂ gas (hydrogenotaxis). Changes in pH were not observed in medium with or without H₂ under the tested growth conditions and times, and these results indicate that the observed taxis was in response to H₂ and not H⁺. Although the commonly used Keller-Segel model was not able to predict the exact response with previously published parameter ranges (values), parameter adjustment allowed the model to replicate the trend on a scale similar to other organisms and chemotactants. The ability to move toward higher concentrations of H₂ could incur an advantage to methanogens that are otherwise outcompeted by those that are able to use H₂ at lower concentrations and/or utilize terminal electron acceptors that are more energetically favorable. The demonstrated chemotactic response would also allow cells to maintain desirable localization with respect to the major energy source as well as allow for proximity to H₂-producers in mixed communities. Thus, hydrogenotaxis could play a crucial role in the establishment and maintenance of microbial interactions at the population- and community-level. The observed hydrogenotaxis could represent a widespread eco-physiological strategy of methanogens and other hydrogen-utilizing microbes that are important to processes such as bioremediation and overall carbon cycling. CH₄ is a potent greenhouse gas that has an estimated global warming potential (GWP) 25–40% higher than CO₂ per molecule. The three largest contributions of CH₄ to atmospheric flux as of 2010 (wetlands, ruminant emissions, and rice cultivation) are the net result of the activity of anaerobic communities dominated by the exchange of H₂. To the best of our knowledge this is the first direct observation of hydrogenotaxis in any domain of life.

**Methods**

**Culturing conditions.** *Methanothermobacter maripaludis* strain S2 was grown in Balch tubes or serum bottles fitted with black butyl stoppers (Geo-Microbial Technologies Inc., Ochelata, OK) and aluminum crimp seals. *Methanothermobacter* Culture Medium (MCC) was prepared under a stream of anoxic 80% N₂, 20% CO₂ and contains per liter 0.33 g KCl, 2.7 g MgCl₂·6H₂O, 3.5 g MgSO₄·7H₂O, 0.14 g CaCl₂·2H₂O, 0.5 g NH₄Cl, 5 g NaHCO₃, 22 g NaCl, 0.14 g K₂HPO₄, 5 mL FeSO₄ solution (0.19 g FeSO₄·7H₂O/100 mL of 10 mM HCl), 1 mL trace metal solution (per 100 mL; 2.1 g Na₂Citrate·2H₂O, adjust pH to 6.5, then 0.45 g MgSO₄·7H₂O, 0.1 g CaCl₂·2H₂O, 0.1 g ZnSO₄·7H₂O, 0.01 g CuSO₄·5H₂O, 0.01 g AlK(SO₄)₂·12H₂O, 0.1 g NiCl₂·6H₂O, 0.2 g Na₂SeO₃, 0.01 g FeCl₃, 0.0033 g Na₂MoO₄·2H₂O) 10 mL of vitamin solution (per liter; 2 mg biotin, 2 mg folic acid, 10 mg pyridoxine HCl, 5 mg thiamine HCl, 5 mg riboflavin, 5 mg nicotinic acid, 5 mg DL-calcium pantothenate, 0.1 mg vitamin B12, 5 mg p-aminobenzoic acid, 5 mg lipoic acid), 1 mL of Resazurin solution (1 g/L). This solution is boiled under a stream of gas before adding 0.5 g cysteine·H₂O/100 mL of 10 mM HCl then cooled under gas. The solution was dispensed anaerobically and autoclaved. After inoculation, the headspace was displaced and pressurized to 25 PSI with anoxic 80% H₂:20% CO₂ through a sterile filter. Modified MCC medium was used for growth experiments with formate, where NaCl was reduced to 10.5 g/L; 200 mM Na-formate and 200 mM 3-(N-morpholino)propanesulfonic acid (MOPS) were added, and no H₂ was added. Cultures for taxis experiments were grown at 30°C (with no shaking) from frozen glycerol stocks in 40 mL MCC in 125 mL serum bottles to late exponential phase, then transferred to 5 mL MCC (10% inoculum) in 18 × 150 mm Balch tubes. These working cultures were grown to stationary phase (around 120 h and average cell density 69,420 cells/mL). 1 mL of culture was added to 5 mL fresh MCC in a Balch tube with no H₂ and either incubated for 4–5 h at 30°C under starvation conditions (no electron source), or used immediately for non-starvation conditions. Media salts were separated from cell suspension prior to taxis experiments by centrifuging the inverted Balch tube (100 × g; 30°C for 10 minutes).

**Electron microscopy.** Pellicles were removed from tubes with a plastic inoculating loop and placed on a round coverslip freshly treated with poly-L-Lysine (1 μg/mL) and fixed in a solution of 2% paraformaldehyde, 2.5% glutaraldehyde and 0.05 M Na-cacodylate overnight at room temperature. Coverslips were rinsed and stepwise dehydrated in ethanol, and then critical point dried on a Samdri-795 (toxinsim, Rockville, MD). Coverslips were mounted on SEM stubs with double-sided carbon tape and colloidal silver, and then sputter coated with Iridium for 35 s at 35 mA. Images were collected on a Zeiss Supra55VP FE-SEM.

**Capillary assay.** Square glass capillary tubes (1.0 mm) with the ends fitted with norprene tubing connected to a polypolypropylene female luer-lock hose barb adapter (Cole-Parmer, Vernon Hills, IL) were partially filled with the cell suspension in an anaerobic chamber that contained only N₂ and CO₂. A 5 mL glass gas-tight luer lock syringe (SGE, Inc., Austin, TX) was used to transfer the cells to the capillary, and left attached to the capillary with the valve closed. A second 5 mL glass gas-tight syringe with 100% H₂ was attached to the gas side of the capillary (see Supplementary Fig. S1 online). The entire syringe/capillary assembly was removed from the anaerobic chamber and placed in a petri dish water bath on the microscope stage and firmly secured with poster putty and tape (see Supplementary Fig. S3 online).

**Microscopic observation of swimming behavior and image analysis.** A Leica TCS SP5 II upright confocal microscope was enclosed in an incubation chamber and was equilibrated to 30°C. High-resolution time lapse images were collected every 0.753 s at the center of the capillary, 0.5 cm from the cell suspension/gas phase interface. A 25 × water-dipping objective was used and a 3 × optical zoom was applied resulting in a final field of view of 206.9 × 206.9 μm² and a pixel size of 0.20 μm. Images were initially acquired for 10 minutes with N₂/CO₂ in the gas phase, then the valve was opened to the H₂ syringe and images were captured for 40 minutes. The control experiments were identical to the above except that 100% Ar was used instead of H₂, and images were acquired for a minimum of 8 minutes before the valve was opened and 38 minutes after.

Images were manually thresholded and binarized using MetaMorph v. 7.6. Binary images were analyzed using Imaris v. 7.5.2 (Bitplane, Inc., South Windsor, CT) with a particle-tracking module (Imaris Track). 1 s, 5 s, and 10 s filter durations were tested where a given track was only analyzed if it was as long or longer than the specified duration. The 5 s filter was used for the described analyses, and differences were not observed in overall trends between track lengths.

**Chemotaxis model.** A one dimensional finite element model was constructed using Comsol Multiphysics Version 4.3a that solves Equations 2 and 4 simultaneously in the liquid domain. Diffusion of hydrogen through the gas domains was modeled by Maxwell-Stefan equations. All model parameters were corrected for temperature (30°C) and salinity (2.65% m/v) of the medium, where possible, and hydrogen consumption rate parameters were estimated from literature values. Supplementary Table S1 online shows all constants used in the model. The geometry of the model consisted of one liquid domain and two gas domains separated by a valve that opens at t = 0 to start the diffusion of hydrogen into the system. The short segment between the valve and the far right boundary is the length of the connection between the valve and the main volume of the gas-tight syringe. The geometry for each experimental replicate was slightly different so average lengths were used. The diffusion of hydrogen through the gas domain of the capillary was
The time steps taken by the solver were allowed to be free with larger steps being taken as step discretization was done with a backward differentiation formula (BDF) method. Hydrogen behind the valve in the second gas domain and a constant population limiting segment between the liquid-gas interface and the point of observation 0.5 cm is expected to be much faster than diffusion through the liquid domain so the precise diffusion of hydrogen into the liquid domain accurately; however, the likelihood of unpredictable factors such as micro-scale mixing due to convection still exists. To provide the flexibility to correct for enhanced mass transport, a correction term was applied such that

$$D_{\text{corr}} = D_{\text{true}} + \delta$$  \hspace{1cm} (5)

Where $\delta$ represents the mass transport enhancement beyond what is predicted from Fickian diffusion alone and $D_{\text{corr}}$ is the corrected diffusion coefficient used in the model for this analysis. Similarly, $\delta$ was also applied to the random cell diffusion coefficient such that

$$\mu' = \mu + \delta$$  \hspace{1cm} (6)

because any correction applied to $D_{\text{corr}}$ would need to be applied to $\mu$ on grounds that convection would affect the movement of cells the same as the pass transport of $H_2$.

**Carbohydrate and protein measurements.** Protein concentrations were determined with the Lowry assay using bovine serum albumin as the standard.\(^{26}\) Hexose sugars were measured by the colorimetric cysteine-sulfuric acid method with glucose as the standard. Pentose sugars were measured with a colorimetric orcinol-FeCl\(_3\) assay with xylene as the standard. A colorimetric carbohydrate assay was used to measure uronic acid concentration with D-galacturonic acid as the standard.\(^{28}\)

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**Acknowledgments**

The authors wish to thank Sara Altenburg for helping with the pH experiment, Betsy Pitts for microscopy assistance, Gill Geseey for encouraging us to pursue the experiments, Al Parker for assistance with statistics, Adam Arkin and Roland Hatzenpichler for helpful comments, and William B. Whitman for his suggestion to use formate. This work was conducted by ENIGMA Ecosystems and Networks Integrated with Genes and Molecular Assemblies (http://enigma.lbl.gov), a Scientific Focus Area Program at Lawrence Berkeley National Laboratory, was supported by the Office of Science, Office of Biological and Environmental Research, of the U. S. Department of Energy under Contract No. DE-AC02-05CH11231. K.A.B. and J.M.C. were also supported by a NSF-IGERT fellowship in the Institute for Global Studies at the University of California (DGSE000892). Partial support for R.G. was provided through the National Science Foundation under CHE-1200632. The confocal microscopy equipment used was purchased with funding from the NSF-Major Research Instrumentation Program and the M.I. Mordock Charitable Trust.
Author contributions
K.A.B. developed experimental design, performed experiments, critically evaluated the model, wrote and revised the manuscript. J.M.C. developed experimental design, performed experiments, created the model, wrote and revised the manuscript. C.D. performed experiments and revised the manuscript. R.G. developed experimental design, critically evaluated the model, and revised the manuscript. M.W.F. developed experimental design, critically evaluated the model, and revised the manuscript.

Additional information
Supplementary information accompanies this paper at http://www.nature.com/scientificreports

Competing financial interests: The authors declare no competing financial interests.

How to cite this article: Brileya, K.A., Connolly, J.M., Downey, C., Gerlach, R. & Fields, M.W. Taxis Toward Hydrogen Gas by Methanococcus maripaludis. Sci. Rep. 3, 3140; DOI:10.1038/srep03140 (2013).

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CORRIGENDUM: Taxis Toward Hydrogen Gas by Methanococcus maripaludis

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The Supplementary Movies that accompany this study were omitted from the original version of this Article.