The chlorosome: a prototype for efficient light harvesting in photosynthesis

Gert T. Oostergetel · Herbert van Amerongen · Egbert J. Boekema

Abstract Three phyla of bacteria include phototrophs that contain unique antenna systems, chlorosomes, as the principal light-harvesting apparatus. Chlorosomes are the largest known supramolecular antenna systems and contain hundreds of thousands of BChl c/d/e molecules enclosed by a single membrane leaflet and a baseplate. The BChl pigments are organized via self-assembly and do not require proteins to provide a scaffold for efficient light harvesting. Their excitation energy flows via a small protein, CsmA embedded in the baseplate to the photosynthetic reaction centres. Chlorosomes allow for photosynthesis at very low light intensities by ultra-rapid transfer of excitations to reaction centres and enable organisms with chlorosomes to live at extraordinarily low light intensities under which no other phototrophic organisms can grow. This article reviews several aspects of chlorosomes: the supramolecular and molecular organizations and the light-harvesting and spectroscopic properties. In addition, it provides some novel information about the organization of the baseplate.

Keywords Chlorosome · Photosynthesis · Electron microscopy · Spectroscopy

Introduction

Since the earliest photosynthetic organisms developed reaction centres, additional peripheral antenna systems have evolved for light harvesting. In these light-harvesting systems, dozens, hundreds or even thousands of (bacterio)chlorophylls can funnel their excitation energy towards reaction centres for charge separation. The green photosynthetic bacteria are anoxygenic phototrophs that contain unique antenna complexes, known as chlorosomes (Blankenship and Matsuura 2003). A chlorosome is actually a kind of organelle. In addition to the green sulphur bacteria (phylum Chlorobi), they are also present in some filamentous anoxygenic phototrophs of the phylum Chloroflexi (formerly known as green non-sulphur bacteria), and in the newly discovered aerobic phototroph, Candidatus Chloracidobacterium thermophilum (Cab. thermophilum) of the phylum Acidobacteria (Bryant et al. 2007). The green sulphur bacteria form the best studied group, and especially Chlorobaculum tepidum (also known as Chlorobium) from the family of Chlorobiaceae, has emerged as a model organism for the group. Within these organisms, the flow of excitation energy goes in the following direction:

Pigments within chlorosomes → CsmA protein in baseplate → FMO protein → reaction center.

Before discussing the structure and function of chlorosomes, some basic facts about the reaction centre and attached proteins are provided. The green sulphur bacterial reaction centre (RC) is a multi-subunit complex, consisting of two copies of the PscA subunit of 82 kDa protein which forms a homodimer, a single copy of PscB (23 kDa), two copies of a 23-kDa subunit named PscC, which in most recent papers is referred to as C-cyt.
(Wen et al. 2008). This gave insight into the structures. Recent chemical labelling and mass spectrometry
of arrii (Fenna and Matthews 1975; Tronrud et al. 1986) and Prosthecochloris aestu-
protein from two species has been determined by X-ray each with a mass of 40 kDa, and the structure of the FMO protein may be involved in stabilization of PscB and/or in the interaction with ferredoxin (see Hauska et al. 2001).

Two copies of the FMO protein trimer associate with the RC and electron microscopy analysis indicated that they are located close to the PscB and PscD subunits (see Hauska et al. 2001). The protein consists of three identical subunits, each with a mass of 40 kDa, and the structure of the FMO protein from two species has been determined by X-ray crystallography. The structures of Prosthecochloris aestu-
arrii (Fenna and Matthews 1975; Tronrud et al. 1986) and Chlorobaculum tepidum (Li et al. 1997) show strong structural similarities. The three monomers form a disc with the symmetry axis perpendicular to the disc plane. There are seven BChl a molecules in a cluster per monomer, and an eighth BChl a molecule has been resolved in newly solved structures. Recent chemical labelling and mass spectrometry data have established the orientation of the FMO protein on the membrane (Wen et al. 2008). This gave insight in the position of the BChls and how these pigments bridge the distance between the baseplate pigments and the core BChl a molecules in the FMO, and how they are involved in efficient excitation energy transfer (Tronrud et al. 2009).

Chlorosomes

Chlorosomes are the largest known antenna structures with some hundreds thousands of bacteriochlorophyll (BChl) c-, d- or e-molecules per chlorosome, which means that there are at least some 5,000 BChls per RC (Hauska et al. 2001). A single chlorosome can contain, depending on the species, about 200,000 to 250,000 BChl c and BChl d molecules, 2,500 BChl a molecules, 20,000 carotenoid molecules, 15,000 chlorobiumquinone molecules, 3,000 menaquin-one-7 molecules, 5,000 protein molecules of 10 different types, and about 20,000 lipid molecules (Bryant and Frigaard 2000). Chlorosomes efficiently capture light and this allows organisms that use chlorosomes for light harvesting to live at extraordinarily low light intensities under which no other phototrophic organisms can grow, exemplified by the findings of species able to survive 100 m below the surface of the Black Sea (Manske et al. 2005).

An interesting property of the chlorosomes is the fact that the majority of the pigments is organized via self-assembly and does not require proteins to provide a scaffold for efficient light harvesting, like the light-harvesting proteins in green plants. This is the major reason why chlorosomes form a source of inspiration for the design of artificial light-harvesting systems. (For a comprehensive review for the self-assembly of chlorins, see Balaban et al. 2005.) In this article, we will review the structural components involved in light harvesting in chlorosomes and their organization. The spectroscopic properties will also be discussed, in relation to the functioning of the chlorosomes and also in relation to the consequences for the structural organization, which after all is still not exactly known.

Supramolecular organization of chlorophylls

Chlorosomes can be considered as elongated sacks, 100–200 nm in length and 40–60 nm in diameter. The overall shape and size of isolated chlorosomes can be easily studied with transmission electron microscopy by classical negative staining with uranyl acetate (Fig. 1). This shows that chlorosomes from different species can differ by at least a factor of 5 in their volume and also vary in shape (Fig. 1, 2). Some are ellipsoid shaped (Fig. 1a), whereas other are conically shaped (Fig. 1b) or irregularly shaped (Fig. 1c). Negative staining has, however, one drawback because it enhances only the contrast of the water-accessible surface; the small negative stain clusters do not penetrate the hydrophobic interior. Cryo-electron microscopy (cryo-EM) of frozen-hydrated samples, on the other hand, gives a total projected density, including the BChl structures. Chlorosomes of C. tepidum, embedded in an amorphous ice layer, give hints of the overall and internal structure. In unstained chlorosomes, a striation pattern is revealed, in a direction parallel to the long axis (Fig. 2a); its calculated diffraction pattern indicates a strong diffraction spot equivalent with a 2.1-nm spacing (inset, Fig. 2a).

Early observations by Staehelin and colleagues indicated that the chlorosome core is separated from the cytoplasm by an approx. 3-nm thick lipid-like envelope layer, which exhibits no substructure (Staehelin et al. 1980). The thickness of the surface layer—the chlorosome envelope—suggests that chlorosomes are surrounded by a lipid monolayer. Since then no further investigations have...
challenged this conclusion. The EM work clearly shows that the observations of Staehelin and co-workers are correct; the borders of the chlorosomes are never thicker than about 2.5–3 nm, which is just a bit more than the 2.1-nm striation pattern (Fig. 2).

The supramolecular organization of the Bchl aggregates within the chlorosomes has been the subject of a long-standing discussion. Early EM observations on thin sections have suggested the presence of 1.2 to 2 nm wide fibrils inside chlorosomes of *Chlorobaculum parvum* (Cohen-Bazire et al. 1964). Based on freeze-fracture electron microscopy, Staehelin et al. (1978, 1980) concluded that Bchl is organized into rod-shaped structures, with a diameter of approx. 5 and 10 nm in *Chloroflexus aurantiacus* and *Chlorobium limicola*, respectively. The *2 nm spacing seen in cryo-electron micrographs of *C. tepidum* chlorosomes (Fig. 2a, 3a), which is also observed by X-ray scattering (Pšencík et al. 2004), seemed at first inconsistent with the results from freeze-fracture EM. Pšencík et al. (2004) interpreted the 2.1-nm spacing as the distance between sheets or lamellae which are oriented parallel to the long axis of the chlorosome. From the extent of the observed striations, it appears that the Bchl sheets are continuous over most of the length of the chlorosomes. The 2-nm spacing remains visible in projection when the chlorosomes are rotated in the microscope about their long axis. This observation led Pšencík et al. to propose a model of an undulating lamellar arrangement of pigment aggregates for three different *Chlorobaculum* species (Pšencík et al. 2004).

Recently, cryo-electron microscopy was performed on intact chlorosomes of *C. tepidum* embedded in a thicker layer of vitreous ice to reveal the arrangement of BChl sheets in wild-type chlorosomes and in chlorosomes from the triple mutant *bchQRU* (Gomez Maqueo Chew et al. 2007), which contains a well-defined >95% homogeneous BChl d (Oostergetel et al. 2007). End-on views of chlorosomes fixed in a vertical position gave a direct clue to the packing of the sheets. They show the presence of multi-lamellar tubules of variable diameter (10–30 nm) with some non-tubular locally curved lamellae in between (Fig. 3). In the *bchQRU* mutant, most chlorosomes contain two tubular domains, as can be deduced from the banding pattern of the 2-nm striations. Overall, the cryo-electron microscopy data show that the *C. tepidum* chlorosomes comprise multi-lamellar tubular domains extending over most of the length of the chlorosome, embedded in a less well-ordered matrix of smaller curved lamellar domains. The notion of multi-walled cylinders is consistent with the results from both freeze-fracture experiments done several decades ago and the more recent cryo-EM observations.

**Molecular organization of chlorophylls**

In addition to the 2-nm lamellar structure, cryo-EM images of *C. tepidum* chlorosomes and their calculated diffraction patterns indicated the presence of a smaller spaced regular structure in the direction of the long axis (Fig. 4). In wild-type chlorosomes, a weak periodicity of 1.25 nm is present (red arrow in Fig. 4b), in the *bchQRU* mutant a relatively strong 0.83 nm regular structure is evident from the diffraction pattern (Fig. 4d) and also directly visible in the image (Fig. 4c, inset). These cryo-EM observations
provide constraints concerning possible packing modes of the BChl molecules in the multi-lamellar tubes.

Due to absence of side chain heterogeneity at C-8 and C-12, limited stereochemical heterogeneity at C-3\(^1\) and absence of a methyl group at C-20 in the \textit{bchQRU} mutant very high resolution magic-angle-spinning (MAS) solid-state NMR data could be obtained. An alternating \textit{syn-anti} ligated BChl \textit{d} stack (Fig. 5a) and an antiparallel monomer stacking model are consistent with the intra-stack distance constraints derived from the NMR data (Ganapathy et al. 2009). When stacks are combined into sheets (Fig. 5b), several inter-stack distances in the antiparallel monomer stacking configuration are larger than those derived from the NMR measurements, whereas the \textit{syn-anti} monomer stack assemblies are consistent with the observed distance constraints.

In chlorophyll aggregates, the \(^1\)H NMR signals shift upfield by ring current effects from neighbouring molecules. Ring current shift calculations were performed for the \textit{syn-anti} monomer stack, the antiparallel monomer model and two earlier structural models that were proposed for BChl \textit{c} in chlorosomes: the monomer-based parallel-stack model (Holzwarth and Schaffner 1994) and the piggy-back dimer model (Egawa et al. 1975). The calculated shifts for the antiparallel monomer stack and the piggy-back dimer configuration were much larger than the experimental shifts. Calculations on the \textit{syn-anti} monomer stack and parallel stack reproduced the experimentally observed shifts. Since the parallel-stack model and the piggy-back dimer model did not satisfy the NMR distance constraints, it was concluded that the \textit{syn-anti} monomer stack was the only model that was consistent with experimental NMR observations and theoretical calculations (Ganapathy et al. 2009).

Based on this \textit{syn-anti} dimer, optimized by molecular mechanics calculations, and the cryo-EM observations, cylindrical models were constructed. For the \textit{bchQRU} mutant, the strong 0.83-nm periodicity in the direction of the long axis (Fig. 4c, d) can be explained by placing the BChl stacks along the circumference of co-axial cylinders, perpendicular to the cylinder axis (Fig. 5, 6). The stacks interconnect via hydrogen bonds which create ultrafast helical exciton delocalization pathways along the BChl cylinders. In the wild-type chlorosomes, the BChl stacks are oriented in the direction of the long axis. Again a
helical O–H···O=C exciton delocalization pathway is present, with opposite handedness as compared to the \( \text{bchQRU} \) mutant. The observed spacing of 1.25 nm (Fig. 4a, b) in this configuration is directly related to the size of a \( \text{syn-anti} \) heterodimer, the basic repeating unit, in the direction of the stack. Simulated projection images from these nanotube models and Fourier analysis confirmed that the supramolecular models were consistent with the experimental data (Fig. 7).

### Organization of the baseplate

The chlorosome baseplate was first described as a 2D paracrystalline structure by freeze-fracture electron microscopy (Staehelin et al. 1980). It may be a monolayer of polar lipids, like the chlorosome envelope. Besides polar lipids, chlorosomes also contain non-polar lipids (waxes) (Sørensen et al. 2008), but their location is completely unknown. About 10 different proteins are embedded in the base plate. Among these, the most abundant is the 59-residue chlorosome protein A (CsmA). The structure of apo-CsmA from \( \text{C. tepidum} \) was determined using NMR spectroscopy (Østergaard Pedersen et al. 2008). Overall, the 59-residue CsmA is predominantly \( \alpha \)-helical in nature with a long helical domain extending from residue 6–36, containing a putative BChl \( \alpha \) binding domain, and a short helix in the C-terminal part extending from residue 41–49. The long N-terminal \( \alpha \)-helical stretch is considered to be immersed into the lipid monolayer confining the chlorosome, whereas the short C-terminal helix is protruding outwards, thus supposedly being available for interaction with the FMO antenna protein. CsmA is known to form stable oligomers in the chlorosome baseplate (Li et al. 2006). In order to assemble two BChl \( \alpha \) molecules in close connection, it was proposed that in the intact baseplate of the \( \text{C. tepidum} \) chlorosomes, CsmA exists as dimers (Østergaard Pedersen et al. 2008), which subsequently can be organized into long rows resulting in the 2D crystalline superstructure of the baseplate as observed by freeze-fracture electron microscopy (Staehelin et al. 1980).

Cryo-EM images of ice-embedded chlorosomes show a large variation of their angular positions. In some specific angular orientation, a thicker line is visible as a kind of a string of beads (Fig. 4a). The strings are considered to be baseplate protein rows in superposition. A calculated diffraction pattern of the part of the chlorosome with the string indicates a repeating distance of 3.3 nm (Fig. 4b).
The baseplates are not directly visible in chlorosomes in an about horizontal position, because the rows have strong overlap with the interior. (Fig. 4c). Diffraction, however, shows again the same distance of 3.3 nm. The fact that the same spacing is observed in two positions is good evidence for the existence of a packing of CsmA molecules in rows with a width of 3.3 nm. A dimer sandwich of CsmA plus BChl $a$ molecules would give such a width. A same conclusion was drawn from observed 3.3 nm spacings for the baseplate of *Chloroflexus aurantiacus* (Psˇencˇı´k et al. 2009).

The positions of spots in diffraction images indicate that the direction of the rows makes an angle of about 40° with the long axis of the chlorosomes in *C. tepidum* but is approximately perpendicular to the long axis in *Cf. aurantiacus*.

Other cryo-EM images hint at a smaller type of spacing, likely of the baseplate. A sharp reflection at 1.1 nm (yellow arrow, Fig. 4) must be caused by a smaller element of the baseplate. As $\alpha$-helices have about this dimension, they are the likely candidates. Psˇencˇı´k and colleagues observed a 0.8-nm spacing in the direction of the long axis in their X-ray scattering profiles (Psˇencˇı´k et al. 2009). Such spacing could be attributed to diffraction from the regular arrangement of CsmA protein in the baseplate as well, although it seems to be too small to originate from a helical packing. Our recent cryo-EM observations do not confirm the 6-nm spacing observed by Staehelin et al. (1980), for which there is no logical explanation either.

**Light-harvesting and spectroscopic properties**

Spectroscopic properties in relation to function

Chlorosomes can contain hundreds of thousands of BChl $c$, $d$ or $e$ (depending on species), which are more closely related to chlorophylls than to bacteriochlorophylls (Blankenship and Matsuura 2003). Monomeric BChl $c$, for instance, has an absorption spectrum that is nearly identical to that of Chl $a$ with maxima around 436 and 668 nm in CCl$_4$ (see, e.g. Olson and Pedersen 1990). Upon
aggregation, the BChl c $Q_y$ absorption maximum shifts to 740–750 nm, very similar to the position of the maximum observed in BChl c containing chlorosomes and aggregates have often been studied as model systems for chlorosomes (see, e.g. Blankenship et al. 1995). Somewhat differently, the absorption maxima of chlorosomes that contain BChl d or e are around 725 and 712 nm, respectively (see, e.g. Blankenship and Matsuura 2003). The strong red shift upon aggregation is due to large excitonic interactions between BChl molecules that are partly stacked upon each other. Many studies have applied excitonic calculations to model and understand the spectroscopic properties of the chlorosomes (see, e.g. Lin et al. 1991; Martiskainen et al. 2009; Prokhorenko et al. 2003; Somsen et al. 1996) and the estimated coupling strengths between nearest-neighbour pigments typically range from $-550$ to $-750$ cm$^{-1}$. These large values lead to delocalization of the excitations over ten(s) of pigments (Prokhorenko et al. 2002; Savikhin et al. 1996, 1998) and they also allow excitations to travel extremely fast throughout the chlorosomes with a “transfer time” of tens of fs between neighbouring pigments as was, for instance, modelled (Prokhorenko et al. 2003). The excitation energy transfer (EET) throughout the chlorosome depends on the overall pigment organization which probably differs for different organisms. EET from bulk BChl c to baseplate BChl a in chlorosomes from *Cf. aurantiacus* occurs for instance within 10 ps (Martiskainen et al. 2009; Savikhin et al. 1996), while EET from bulk BChl c to baseplate BChl a in chlorosomes from *Chlorobium phaeobacteriodes* is approximately 10 times as slow (Pšenčík et al. 2003).

The large coupling strengths are reminiscent of those in J-aggregates but in that case they lead at the same time to substantial narrowing of the absorption bands (see, e.g. Fidder and Wiersma 1991). This is unfavourable for light-harvesting because this implies that only light in a very narrow wavelength region can be absorbed. However, the absorption bands of chlorosomes are rather broad which is at least partly due to the fact that the BChl c/d/e composition in vivo consists of a mixture of many homologues (Gomez Maqueo Chew et al. 2007; Olson and Pedersen 1990), which leads to structural disorder and thus to spectral broadening (see also Prokhorenko et al. 2003 Somsen et al. 1996).

It is worthwhile to point out that the efficiency of EET to a RC is apart from the rate of EET and the number of pigments also determined by the ratio of the number of pigments in “contact” with the RC and the total amount of pigments. Suppose, for instance, that there would be 10 out of $10^5$ BChls in close contact to an RC ($N_{\text{transfer}} = 10$, $N_{\text{total}} = 10^5$) and that the EET time from any of these 10 pigments to the RC would be 1 ps. Even if the energy transfer between the BChl c molecules would be infinitely fast, the overall transfer time would be $N_{\text{total}}/N_{\text{transfer}}$ times $1 \text{ ps} = 10 \text{ ns}$, because the probability for excitations to be on a BChl c next to the RC would be $N_{\text{transfer}}/N_{\text{total}}$, thereby lowering the effective transfer time to the RC with a factor of $10^7$ and also the transfer efficiency because of competing loss processes (fluorescence, internal conversion and intersystem crossing). The presence of a baseplate that contains BChl a molecules which absorb at longer wavelengths (790–800 nm) improves the efficiency of transfer to the RC (or in fact FMO protein/B808–866) substantially. Chlorosomes from *Chlororflexaceae* typically have a ratio BChl c:BChl a of 50 (Blankenship and Matsuura 2003), and the relatively large amount of BChl a with excited-state energy levels that are significantly below those of BChl c leads to fast excited-state population within the baseplate (610 ps, see also above). Transfer from baseplate to RC is a factor of $\sim 50$ faster than it would have been from BChl c purely for entropic reasons because $N_{\text{total}}/N_{\text{transfer}}$ is a factor of 50 smaller for BChl a as compared to BChl c. Of course, this is a simplified view because also other factors play a role like overlap of donor emission and acceptor absorption spectra and relative orientations of the transition dipole moments. By increasing the number of BChl a molecules in the baseplate, the rate of extracting excitations from the BChl c pool will increase (also for entropic reasons) but on the other hand it will decrease the transfer to the RC because of lowering the ratio $N_{\text{total}}/N_{\text{transfer}}$. It is clear that the ratio of BChl c to BChl a is an important parameter for determining the

Figure 7 Cylindrical model of the packing of concentric lamellae in the *Chlorobaculum tepidum* bchQRU mutant, based on distances as observed by electron microscopy and solid-state NMR spectroscopy (Ganapathy et al. 2009). The spacing between layers is 2.1 nm. The green band indicates the position of individual Bchl molecules in four stacks of syn-anti dimers. In the wild-type chlorosomes, the stacks run in the direction of the cylinder axis.
The third category of pigments in chlorosomes is the one of the carotenoids, constituting ~8% of the total amount of pigments in *chlororflexaceae* and ~4% in *chlorobiaceae* (Blankenship and Matsuura 2003). They transfer excitation energy to the BChls and, for instance, in *Cf. aurantiacus* a transfer efficiency to BChl *c* of 65% was reported (Van Dorssen et al. 1986), implying that at least 65% of the carotenoids should be in Van der Waals contact with BChl *c*. Direct interactions between BChls and carotenoids have also been inferred from changes in the BChl Stark spectrum (Frese et al. 1997) and the BChl absorption spectrum in the absence of carotenoids (Arellano et al. 2000; Kim et al. 2007). On the other hand, the carotenoids also protect chlorosomes against photodegradation and it was found that carotenoid-free chlorosomes photodegrade approximately three times faster than wild-type ones (Kim et al. 2007). However, no proof for BChl *c* triplet quenching by carotenoids could be found in *Cf. aurantiacus* and *C. tepidum* (Carbonera et al. 2001), whereas Arellano and coworkers found evidence for BChl *a* triplet quenching by carotenoids but not for BChl *e* triplet quenching in *Chlorobium phaeobacteroides* strain CL1401 (Arellano et al. 2000). Triplet quenching of (B)Chls by nearby carotenoids is usually occurring in photosynthetic light-harvesting systems to avoid the formation of deleterious singlet oxygen. It was demonstrated that monomeric BChl *e* sensitizes singlet oxygen formation but aggregated BChl *e* does not (Arellano et al. 2002). In line with these results, Kim and colleagues studied a carotenoid-free mutant of BChl *c* containing *C. tepidum* and found that a significant fraction of the BChls forms a long-lived, triplet-like state that does not interact with oxygen and it was proposed that these states are triplet excitons formed by triplet–triplet interaction between BChls that are lower in energy than the singlet oxygen state (but also than the triplet energy level of carotenoids) (Kim et al. 2007).

Light spectroscopy and structure

The large excitonic red shift of the chlorosomes requires an arrangement of the pigments that is reminiscent of the organization in J-aggregates (Moll et al. 1995), i.e. head-to-tail or head-to-head organization and many possibilities have been provided in literature over the years (for an “early” overview see, for instance, Blankenship et al. 1995). Most of these proposed aggregates were linear but to account for the relatively pronounced circular dichroism (CD) helical and cylindrical models were introduced (Lin et al. 1991; Prokhorenko et al. 2003; Somsen et al. 1996; Linnanto and Korppi-Tommola 2008) in which the J-type organization was kept intact. Over the years also many linear-dichroism (LD) measurements have been performed and these all demonstrated that the transition dipole moment corresponding to the long-wavelength *Q* transition dipoles make a relatively small angle with the long axis of the chlorosomes (for more details see below). Also polarized transient absorption measurements (Lin et al. 1991; Pšenčík et al. 2003) and polarized fluorescence measurements on non-oriented chlorosomes (Ma et al. 1996; Van Dorssen et al. 1986) and chlorosomes in intact cells of *C. limicola* (Fetisova et al. 1988) indicated a high degree of ordering, that was more or less consistent with the LD results.

As LD measurements provide spectroscopic information that may be used to verify structural models we will briefly address the LD of chlorosomes. The LD (*AA*) is defined as the difference in absorption (*A*) of light polarized parallel (*v*) and perpendicular (*h*) to the orientation axis of the sample (expansion direction of a squeezed gel containing the chlorosomes or the direction of an orienting electric field): *ΔA* = *A* *v* − *A* *h* (see also Garab and Van Amerongen 2009). LD measurements provide the angle *θ* between a transition dipole moment and the long axis of the chlorosome. Values between 15° and 27° were obtained for the transition dipole moment of the main *Q* transition band and the long axis of the chlorosomes from *Cf. aurantiacus* (Frese et al. 1997; Griebenow et al. 1991; Matsuura et al. 1993; Van Amerongen et al. 1988, Van Amerongen et al. 1991). Single molecule experiments on chlorosomes from *Cf. aurantiacus* also showed preferential orientation of the *Q* transition dipole moment along the long axis, and from these results an average angle of around 29° can be inferred. Recent experiments on chlorosomes from *C. limicola* are somewhat different but also show a high degree of ordering (Shibata et al. 2009).

It should be noted that the orientation of the main *Q* transition dipole is not necessarily parallel to the transition dipole moment of the individual BChls. For ideal helical and cylindrical models in which the broadening of the absorption bands is ignored, the red-most band is parallel to the helix/cylinder axis (positive LD), whereas degenerate perpendicular components absorb more to the blue (Lin et al. 1991; Somsen et al. 1996), creating negative LD. However, when homogeneous and inhomogeneous broadenings of the absorption bands are also included, the picture is less extreme and the reduced LD decreases more gradually upon going to the blue. Such a decrease has indeed been reported (Griebenow et al. 1991; Matsuura et al. 1993). Earlier polarized transient absorption measurements showed a decrease in anisotropy upon going to the blue and it was explained in a similar way (Lin et al. 1991).
Although the angle reported above refers to the transition dipole moments of excitonic transitions, the orientations of the transition dipole moments of the individual pigments can be obtained in a straightforward way. In fact, if one integrates the LD over the entire \( Q_y \) band and compares it to the integrated absorption, one obtains the angle of the transition dipole moment of the individual BCHls with respect to the long axis of the chlorosomes (for the background theory we refer to (Somsen et al. 1996; Van Amerongen et al. 2000), but the underlying reason is that excitonic interactions shift absorption bands but do not alter the total amount of dipole strength along a particular axis). Although it has never been explicitly calculated in literature, it can easily be done from the available data and it appears that the obtained angle for the individual pigments is at most a few degrees larger than the one of the main (excitonic) absorption band. Thus, it is concluded that the above-mentioned results on chlorosomes from \( Cf. aurantiacus \) demonstrate that the angle between the \( Q_y \) transition dipole moment of the individual BCHl c molecules is 25° ± 6° with respect to the long axis, where the error reflects the spread in the reported values. These numbers can be taken into account when building molecular models (Prokhorenko et al. 2003).

There is a remarkable variability in the shape of the CD spectra that have been reported in literature. This variability was even present for chlorosomes that were prepared in an identical way, whereas the absorption and linear-dichroism spectra were identical. It was demonstrated in (Somsen et al. 1996) that a slight reorganization of cylindrical aggregates could explain these results, but later it was demonstrated that the variability in CD could elegantly be explained by variations in the length of the cylindrical aggregates (which do not substantially affect the absorption and LD spectra (Didraga et al. 2002; Didraga and Knoester 2003; Prokhorenko et al. 2003).

In the recently proposed model for chlorosomes from a triple mutant of \( C. tepidum \), the \( Y \)-axis of BCHl c along which the \( Q_y \) transition dipole moment is oriented makes an angle of 55° with the local cylinder axis (Ganapathy et al. 2009). This means that the LD integrated over the \( Q_y \) band should be very close to zero. Due to exciton coupling, the LD is again expected to be positive on the long-wavelength side and to keep the integrated LD close to zero this should then be compensated by negative LD on the short-wavelength side. Linear-dichroism spectra of these particular chlorosomes have not been presented in literature.

There is one more issue that should be clarified and this concerns the Stark spectrum of chlorosomes. Chls and BCHls possess a difference dipole moment \( \Delta \mu \) between ground and excited \( (Q_y) \) state that is responsible for a feature in the Stark spectrum with the shape of the second-derivative of the absorption spectrum (see, e.g. Boxer, 2009). The intensity of this contribution is a measure for the value of \( \Delta \mu \). Remarkably, in contrast to all the known Stark spectra of photosynthetic complexes, there is no such feature for the \( Q_y \) absorption band and \( \Delta \mu \) is equal to 0 (Frese et al. 1997). This has been explained by an antiparallel organization of strongly coupled BCHl c molecules in the chlorosome, either because of antiparallel-dimer building blocks or because of the presence of antiparallel linear stacks. Such an antiparallel organization is not present in the model for the chlorosomes of triple mutant of \( C. tepidum \) mentioned above (Ganapathy et al. 2009).

Therefore, it is expected that Stark measurements on these chlorosomes will show second-derivative character in the \( Q_y \) region and together with the LD measurements they might form another way of testing the current model. Finally, it is worthwhile to point out that the lamellar model that was proposed by Pšenčík et al. (2004) cannot explain the pronounced CD spectra of chlorosomes (Linnanto and Korppi-Tommola 2008) although the authors could not rule out the simultaneous presence of lamellar and cylindrical structures. According to the most recent EM data presented above, such a coexistence seems indeed to be the case (Ganapathy et al. 2009; Oostergetel et al. 2007).

Closing remarks

In conclusion, chlorosomes are fascinating organelles because of their amazing capacity of light harvesting. The need for harvesting a broad range of the spectrum of light constrains the composition of the BCHl molecules and the amount of order in the packing, for which now a consistent model is available. This model describes the molecular and supramolecular packing and can be further tested, for instance, with LD, which will provide useful information on the long-range ordering of the pigments. An intriguing feature is the thin envelope which consists of only one membrane leaflet, an uncommon phenomenon in nature. It nevertheless contains 10 different proteins, and some of them play important roles in determining chlorosome size as well as the assembly and supramolecular organization of the BCHl c aggregates within the chlorosome (Li and Bryant 2009). It shows the complicated fine tuning of the participating components. As the determined interactions between the BCHls, however, are rather simple it may one day be possible to build artificial photosynthetic chlorosome-based systems that efficiently convert solar energy to electricity or fuel.

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