Bioactivities and Structure–Activity Relationships of Natural Tetrahydroanthraquinone Compounds: A Review

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Tetrahydroanthraquinones are a kind of important microbial secondary metabolites with promising biological activities. Most of them were found in microorganisms, a few were derived from Chinese herbal medicine. In this review, aiming to provide basis for the further research and development of tetrahydroanthraquinone compounds, we summarized the physiological activities of natural tetrahydroanthraquinone compounds, including anticancer, anti-microbial, and antidiabetic activities. The source, structure, and action mechanisms of active tetrahydroanthraquinones are described in detail. Furthermore, this review firstly analyzed the structure–activity relationship of tetrahydroanthraquinones. Our study will serve as a valuable guideline for further research on the structural optimization, mechanism study, and development of tetrahydroanthraquinone as novel drugs. Aiming to provide references for further studies and development of tetrahydroanthraquinone compounds.

Keywords: tetrahydroanthraquinone, anti-cancer, structure–activity relationship, bioactivities, altersolanol A, bostrycin

INTRODUCTION

Anthraquinones characterized by an anthraquinone scaffold structure, are widely distributed in plant as secondary metabolites. Modern pharmacological researcher showed that anthraquinones have various potent activities, including anticancer, anti-inflammatory, anti-injury, antibacterial, anti-osteoporosis, antioxidant, etc. (Li and Jiang, 2018).

Tetrahydroanthraquinone is a class of derivatives of anthraquinone in which the unsaturated double-bonds on the benzene ring A are hydrogenated by four hydrogens (Figure 1). Most of them were found in microorganisms, while a little were derived from Chinese herbal medicine. To date, about 60 different tetrahydroanthraquinones are found, of which only nine are from plants (Wang C. et al., 2015). Because of some of tetrahydroanthraquinones showing good pharmacological activity, such as anticancer, antibacterial, anti-malarial, and anti-viral, more and more researchers...
are being focused on the exploration of their pharmacological activity, mechanism, and structure activity relationships.

Structurally, tetrahydroanthraquinones are generally classified into tetrahydro-9,10-anthraquinones, hydroxyphenanthrenes, tetrahydro-1,4-anthraquinones, and bi-tetrahydroanthraquinone, and according to their sources, tetrahydroanthraquinones can be divided into the ones from microorganisms and the ones from plants. In this paper, we focus on the pharmacological activities and related structure–activity relationships of active tetrahydroanthraquinones, aiming to provide references for further studies and development of tetrahydroanthraquinone compounds.

**STRUCTURE AND CLASSIFICATION**

**Tetrahydroanthraquinones From Microorganisms**

**Tetrahydro-9,10-Anthraquinones**

The tetrahydroanthraquinone ring is the fundamental parent nucleus of tetrahydroanthraquinones (Figure 1). Hydroxyls could locate in the C-1 to C-8 positions. In general, a compound with single tetrahydroanthraquinone nucleus is called a tetrahydro-9,10-anthraquinone. There are nineteen tetrahydro-9,10-anthraquinones that were isolated from endophytes and marine strains (Table 1 and Figure 2). Besides, nine kinds of tetrahydro-9,10-anthraquinones extracted from the root of *Prismatomeris connata*, and they are named as prisconnatanones A-I (Table 1 and Figure 2). Most tetrahydro-9,10-anthraquinones isolated from microorganisms have a structure with 5-OH and 7-OCH3, and substituent groups.
at A ring can be variable. However, tetrahydro-9,10-anthraquinones extracted from \textit{P. connata} have a basic structure with 2-OH (R) and 3-CH\textsubscript{3} (R), and their C ring mostly substituted with hydroxyl and methoxy groups.

**Hydroxyphenanthrenes**  
Hydroxyphenanthrene is a class of compounds in which the C-9 or C-10 position is hydrogenated to a hydroxyl group compared to a tetrahydroanthraquinone, and generally named as 10-hydroxy-1,3,4,4a,9a,10-hexahydroanthracene-9(2H)-one. There are ten hydroxyphenanthrenes that has been isolated and identified. Their detailed source and structure information were displayed in Table 2 and Figure 3.

**Tetrahydro-5,8-Anthraquinones**  
Tetrahydro-5,8-anthraquinones is a class of compounds in which...
the C5 and C8 positions of a tetrahydroanthraquinone are oxidized to a ketone group. They are rare in natural products due to their instability. In nature, most of anthraquinones and tetrahydroanthraquinones exist in the form of C9 and C10 ketone groups. Although rare, five tetrahydro-5,8-anthraquinones have been isolated and identified as (1S,3R)-austrocortirubin (6-methoxy-1β,3β,9,10-tetrahydroxy-3α,3β-dimethylanthracene-9,10-dione), (1S,3S)-austrocortirubin (6-methoxy-1β,9,10-trihydroxy-3α,3β-dimethylanthracene-9,10-dione), 1-deoxyaustrocortirubin (7-methoxy-2β,9,10-trihydroxy-2α-methylanthracene-9,10-dione), deoxybostrycin (7-methoxy-2β,3β,9,10-tetrahydroxy-2α-methylanthracene-9,10-dione), and bostrycin (7-methoxy-1α,2β,3β,9,10-pentahydroxy-3α,3β-dimethylanthracene-9,10-dione). The detailed source and structure information was displayed in Table 3 and Figure 4.

Bi-Tetrahydroanthraquinones
Bi-tetrahydroanthraquinones were formed by two tetrahydroanthraquinones or a tetrahydroanthraquinone and an anthraquinone through dehydration condensation to establish dimeric alterporriols. Seventeen bi-tetrahydroanthraquinones have been reported (Table 4 and Figure 5), among which based on the type of the monomeric units, biaryl linkage, presence of axial chirality, they could be further classified as anthraquinone connected to tetrahydroanthraquinone unit via C-5–C-5′ biaryl linkage (44, 45, 51, 55, and 56), and C-7–C-5′ biaryl linkage (48, 49, 59, and 60). As to the bi-tetrahydroanthraquinone units, compounds 46, 47, 50, 57, and 58 coupled by C-5–C-5′ biaryl linkage, compounds 53 and 54 presented a C-7–C-5′ biaryl linkage, only compound 52 featured a C-4–C-4′ cyclohexene connection. Moreover, compounds 50, 52, 53, and 54 demonstrated as axial

TABLE 2 | The source and activities of hydroxyphenanthrene compounds.

| No | Name               | Source                       | Activities                      | Reference               |
|----|--------------------|------------------------------|--------------------------------|-------------------------|
| 29 | Altersolanol J     | NRRL 29097                   | –                              | (Höller et al., 2002)  |
| 30 | Altersolanol K     | Stemphylium globiferum       | –                              | (Debbab et al., 2009)  |
| 31 | Altersolanol L     | S. globiferum                | –                              | (Debbab et al., 2009)  |
| 32 | Tetrahydroaltersolanol B | Phomopsis sp. PSU-MA214 | –                              | (Klaiklay et al., 2012) |
| 33 | Tetrahydroaltersolanol C | Phomopsis sp. PSU-MA214 | Antiviral activity             | (Klaiklay et al., 2012; Zhang S.L. et al., 2016) |
| 34 | Ampelanol          | Ampelomyces sp.              | –                              | (Hy et al., 2008)      |
| 35 | Xylanthaquinone    | Xylaria sp. 2508             | –                              | (Huang et al., 2014)   |
| 36 | 2-O-acetyltersolanol L | Stemphylium sp. 33231       | –                              | (Zhou et al., 2014)    |
| 37 | Altersolanol Q     | Stemphylium globiferum       | –                              | (Moussa et al., 2016)  |
| 38 | 10-methytersolanol Q | S. globiferum               | –                              | (Moussa et al., 2016)  |

The structure of hydroxyphenanthrene compounds (29–38).
chirality monomers, compounds 44 and 45, compounds 46 and 47, compounds 48 and 49 were atropodia stereomers.

PHARMACOLOGICAL EFFECTS

Anti-Tumor Activity of Tetrahydro-9,10-Anthraquinones

Some natural anthraquinones have been proven to have anti-tumor effects, such as rhein and emodin (Li and Jiang, 2018; Han et al., 2018). Tetrahydroanthraquinones, a class of anthraquinones, can also exhibit anti-cancer activity such as inhibiting cell proliferation, invasion, metastasis, and angiogenesis by inducing cell apoptosis, arresting cell cycle or suppressing the relevant enzymes. Altersolanol A, one of the most researched tetrahydroanthraquinone, exhibits antitumor activities against broad spectrum cancers (bladder, colon, gastric cancers, etc.) (Zhang N. et al., 2016). It can inhibit the proliferation and migration of both adherent cells K562 and non-adherent cells A549, whereas leave non-cancer cells (PBMCs cells) unaffected (Teiten et al., 2013). The anti-tumor activity of Altersolanol A is correlated with its pro-apoptotic and anti-invasive effect due to the inhibition of NF-κB transcription (Teiten et al., 2013). Another study also showed Altersolanol A exerted anti-cancer activity by inhibiting angiogenesis at low concentration in vitro and in vivo, and suppress the proliferation, tube formation, and migration of endothelial cells (Phunlap et al.,

| TABLE 3 | The source and activities of tetrahydro-5,8-anthraquinones compounds. |
|---|---|---|---|
| No | Name | Source | Activities |
| 39 | (1S,3R)-austrocortirubin | Dermocybe splendida | – |
| 40 | (1S,3S)-austrocortirubin | Cortinarius sp | Antitumor activity |
| 41 | 1-deoxyaustrocortirubin | Cortinarius sp | – |
| 42 | Deoxybostrycin | Xylaria sp. 2508 | Antitumor activity; Antibacterial activity; Antimalarial activity |
| 43 | Bostrycin | Xylaria sp. 2508 | Antitumor activity; Antibacterial activity |

| TABLE 4 | The source and activities of Bi-tetrahydroanthraquinones compounds. |
|---|---|---|---|
| No | Name | Source | Activities |
| 44 | Alterporriol A | Stemphylium sp. 33231 | – |
| 45 | Alterporriol B | Stemphylium sp. 33231 | – |
| 46 | Alterporriol D | Pleospora herbarum | – |
| 47 | Alterporriol E | P. herbarum | – |
| 48 | Alterporriol G | Stemphylium globuliferum | Antibacterial activity |
| 49 | Alterporriol H | S. globuliferum | – |
| 50 | Alterporriol F | Pleospora herbarum | – |
| 51 | Alterporriol N | Alternaria sp. XZSBG-1 | – |
| 52 | Alterporriol S | Alternaria sp. XZSBG-1 | – |
| 53 | Alterporriol T | Stemphylium sp. 33231 | – |
| 54 | Alterporriol U | Stemphylium sp. 33231 | – |
| 55 | Alterporriol V | Alternaria sp. XZSBG-1 | – |
| 56 | Alterporriol W | Stemphylium sp. 33231 | – |
| 57 | Aectylalterporriol D | S. globuliferum | – |
| 58 | Aectylalterporriol E | S. globuliferum | – |
| 59 | Alterporriol X | S. globuliferum | Antitumor activity |
| 60 | Alterporriol L | Alternaria sp. ZJ9-6B | – |
FIGURE 5 | The name of bi-tetrahydroanthraquinones compounds (44–60).
inhibited the growth of these same cancer cell lines with IC_{50} values of 2.64, 5.39, 5.90, 4.19, 6.13, and 6.68 μg/ml, respectively (Xuekui et al., 2011). For bostrycin, previous study reported it inhibited cell proliferation via upregulation of miRNA-638 and miRNA-923 and downregulation of the PI3K/Akt pathway (Wei-Sheng et al., 2011). Besides, it could induce apoptosis and cell cycle arrest in A549 cells (Wei-Sheng et al., 2011). Further study uncovered that bostrycin inhibits the proliferation of breast cancer cells through changing the structure of PTPIB (protein tyrosine phosphatase 1B) and inhibiting its activity (Dongni et al., 2013).

To further explore the structure–activity relationship of bostrycin, Hong Chen et al. (2012) synthesized 18 bostrycin derivatives through structural modification at positions 2, 3, 6, and 7. And found that dioxy carbonyl groups at C-2 and C-3 positions, tertiary amino groups at C-6 position and alkylthio groups at C-6 and C-7 positions of the bostrycin could enhance cytotoxicity of bostrycin derivatives.

**Anti-Tumor Activity of Bi-Tetrahydroanthraquinones**

Alterporriol L could effectively inhibit the proliferation and growth of breast cancer cell line MCF-7 (IC_{50}, 13.11 μM) and MDA-MB-453 cells (IC_{50}, 20.04 μM), and there was a dose-dependent manner of cell death. Moreover, alterporriol L could induce cancer cell apoptosis and necrosis through triggering the generation of oxidative stress (Huang et al., 2012).

It is interesting that we have not found any report of hydroxyp henanthrenes with anti-tumor activity. This may suggest that the tetrahydroanthraquinone skeleton with carbonyl groups at C-9 and C-10 positions is important for the anti-tumor activity of tetrahydroanthraquinones.

**Antimicrobial Activity**

Many tetrahydroanthraquinones exhibit good antimicrobial activities. Altersolanols A–C and E inhibited the growth of all Gram-positive bacteria and Pseudomonas aeruginosa IF0 3080 with minimum inhibitory concentration (MIC) value ranging from 12.5 to 25 μg/ml. However, Altersolanol D and F, lack of hydroxyl group at C-5 compared to Altersolanols A–C and E, even at concentration as high as 100 μg/ml, have not inhibited bacteria growth. This indicated that the hydroxyl group at C-5 position was necessary to the antibacterial activities of tetrahydroanthraquinones (Yagi et al., 1993). Coniothranthraquinone 1 showed antibacterial activity against Staphylococcus aureus ATCC 25923 (SA) and Staphylococcus aureus SK1 (MRSA), with MIC values of 16 and 8 μg/ml, respectively (Khamthong et al., 2012; Ng et al., 2015). While trichodermaquinone showed antibacterial activity against MRSA, with a MIC value of 200 μg/ml, had no inhibition on SA (Khamthong et al., 2012). This study suggested that the hydroxyl group at C-5 position was necessary to the antibacterial activities of tetrahydroanthraquinones.

Deoxybostrycin and Bostrycin showed strong antimicrobial activities against Staphylococcus aureus, Escherichia coli, Pseudomonas aeruginosa, Sarcina ventriculi, Bacillus subtilis with an IC_{50} of 3.13 μg/ml, and inhibited Candida albicans with an IC_{50} of 12.5 μg/ml. Besides, Deoxybostrycin showed good anti-
mycobacterial activity, it exhibited a better inhibitory effect on clinical multidrug-resistant *M. tuberculosis* (K2903531 and 0907961) than the first line anti-tuberculosis drug (Nigrosporin) (Yagi et al., 1993). Alterporriol G showed antibacterial activity only against *Streptomyces pneumonia* (Aly et al., 2008).

**Antiviral Activity**

Tetrahydroaltersolanol C exhibited a significant anti-PRRSV (Porcine reproductive and respiratory syndrome virus) activity with a EC_{50} value of 12.11 µM, it inhibited the internalization and replication of PRRSV, but did not directly inactivate the virus or block its adsorption to cell surface (Zhang S.L. et al., 2016).

**Antidiabetic Activity**

4-des-hydroxyl altersolanol A significantly reduced the level of blood sugar in alloxan-induced diabetic mice (Uzor et al., 2017).

**CONCLUSIONS AND PERSPECTIVES**

Anthraquinone compounds and their natural derivatives, especially tetrahydroanthraquinones, showed a considerably wide range of pharmacological effects, and 60 tetrahydroanthraquinones have been found since altersolanol A was originally reported in 1967 (Stoessl, 1969b). Some of them exhibited considerable cytotoxicity, antimicrobial, antiviral activity, and hypoglycemic activities. In this review, anti-tumor, anti-microbial, antiviral activity, and anti-diabetic activities of tetrahydroanthraquinones are summarized in detail, and 17 active ones are involved. We try to sum up the structure and activity relationship of tetrahydroanthraquinones from previous literatures.

Anthraquinones exhibited potent antitumor effect, while they also with DNA toxicity, can be inserted into the helical structure of DNA in the form of a flat structure, affecting the transcription and DNA replication (Adhikari and Mahar, 2016). Hence, anthraquinones showed stronger toxicity than pharmacological effect. Tetrahydroanthraquinones, especially tetrahydro-9,10-anthraquinones, avoided DNA toxicity caused by anthraquinone planar structure. Tetrahydroanthraquinone forms a dimensional construct of cyclohexene after hydrogenation, and forms two or more chiral centers by substitution with OH or CH₃. This increased the potential druggability and formed multiple targeting centers, providing more possibilities for chemical modification and structure transformation. There are few reports in the available literature that the antitumor effect of tetrahydroanthraquinone is related to DNA toxicity, except for (15,3S)-astrocoritubin (Wang Y. et al., 2015). More studies suggested that the antitumor effect of tetrahydroanthraquinone is through targeting signaling pathways, including NF-κB, PI3K/Akt pathway (Xie et al., 2010; Zhu et al., 2012; Genov et al., 2016). Suggesting that the antitumor mechanisms of tetrahydroanthraquinones are different from that of anthraquinone. Tetrahydroanthraquinones deserve more attention and more research.

For anti-cancer activity, it seems that the p-quinone moiety of tetrahydroanthraquinone is fundamental, as reduction of one of the carbonyl groups of the quinone moiety nullified the cytotoxicity of the tetrahydroanthraquinone derivatives (Zhang N. et al., 2016). Adding short side chains to the benzoquinone increases cytotoxicity of tetrahydroanthraquinones, whose cyclohexyl ring is substituted with two hydroxyl groups with the appropriate stereochemistry, and elimination of both or even a single hydroxyl, or change of stereochemistry of the tertiary hydroxyl eliminates biological activity, and additive hydroxyl moiety at C-1 of a tetrahydroanthraquinone might be the active profile for inhibiting lung tumor cell growth. Linker at a 3-atom with a phenyl or para-chlorophenyl moiety also can enhance cytotoxicity (Phunlap et al., 2013; Teiten et al., 2013; Li and Jiang, 2018). For anti-microbial activity, from the few studies, the hydroxyl group on the C-5 might be crucial to the anti-microbial activity (Yagi et al., 1993).

The most pharmacological researches of tetrahydroanthraquinones focused on cytotoxicity and antitumor mechanisms, suggesting its potential for developing antitumor drugs. While we found some of them are not suitable for drug research. Biterahydroanthraquinones are not suitable for drug research due to large weight and large steric resistance. Tetrahydro-5,8-anthraquinones are a class of rare compounds with unstable structure, with tautomerism at the 9, 10 positions of its 5,8-dione. They are not suitable for drug development due to the difficult in structural modification and pharmacological research. Hydroxyphenanthrenes is characteristic by the C9 carbonyl group undergoes a reduction reaction to form a hydroxy substitution. However, under acidic conditions, the C9 carbonyl group undergoes dehydration reaction, and becomes keto group. therefore, special attention needs to be paid to the conditions in pharmacological and synthetic research of Hydroxyphenanthrenes.

The tetrahydroanthraquinone isolated and identified are mainly from endophytes, and some of them are isolated from marine fungi and plants. Some pharmacological activities of tetrahydroanthraquinones are reported, yet are not enough. There is still a strong possibility that some tetrahydroanthraquinones with better regulatory activities remain in the shadow or have not been fully studied. Our study will serve as a valuable guideline for further research on the structural optimization, mechanism study, and development of tetrahydroanthraquinone as novel drugs.

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**Conflict of Interest:** The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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