Analysis of Syngas Results of the Maindepth Coal Gasification Process with Gasification Downraft Methods

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ABSTRACT

According to the Geological Agency (2015), Indonesia's resources total are 106.845 billion tons and coal reserves are 32.263 billion tons. Behind the reasons for choosing coal as an energy source, there are still some disadvantages of using coal directly. One of them is that coal contains a lot of pollutants that are harmful to the environment. Coal releases gases (CO2, N2O, NOx, SOx and Hg) that caused global warming and pollution. The process of solid convert to gas is called Gasification is. In contrast to combustion, the process of is a breaking of the carbon chain into the form of other elements or chemical compounds. In this study, the gasification process was carried out using the downdraft method. Gasification of coal will be produce producer gas in the form of synthetic gas (syngas) with the main components consisting of CO, H2, CO2 and N2 and low in pollutants. This study aims to determine the effect of the type of coal seam on the syngas produced, the calorific value of the syngas, and the effect of the air flow rate.

Keywords: Gas Fuel, Downdraft Gasification, Coal, Syngas Results

1. INTRODUCTION

Indonesia has potential resources and coal reserves which are spread mostly on the islands of Kalimantan and Sumatra, and a small portion of the rest is spread across several locations on the islands of Java, Sulawesi and Papua. According to the Geological Agency (2015), Indonesia's total resources are 106.845 billion tons and coal reserves are 32.263 billion tons. The quality of Indonesian coal resources is quite varied in terms of caloric parameters, ash content, sulfur content, total moisture, and other parameters (BAPPENAS, 2019). Behind the reasons for choosing coal as an energy source, there are still some disadvantages of using coal directly. One of them is that coal contains a lot of pollutants that are harmful to the environment. Coal releases gases (CO2, N2O, NOx, SOx and Hg) that cause global warming and pollution.

The development of coal conversion in Indonesia is basically an inseparable part of encouraging the increase in added value of coal that must be carried out by coal entrepreneurs as stipulated in Government Regulation Number 23 of 2010 articles 94, 95, and 96 and national energy policies based on Government Regulations of the Republic of Indonesia. Number 79 of 2014 concerning energy diversification (article 18 paragraph 2 point b) states that one of the energy diversification is through increasing the use of low quality coal for gasified coal.

Gasification is a process of converting solid fuel into gas. In contrast to combustion, the gasification process is the process of breaking the carbon chain into the form of other elements or chemical compounds. The gasification process requires little oxygen and water vapor is often used for the combustion process [2]). By converting coal into gas, unwanted materials contained in coal such as sulfur compounds, carbon dioxide (CO2), and ash can be removed from the gas using certain methods so that clean gas can be produced. Coal gasification will produce producer gas in the form of synthetic gas (syngas) with the main components consisting of carbon monoxide (CO), hydrogen (H2), carbon dioxide (CO2) and nitrogen (N2) which are low in pollutants. So that energy experts have focused on developing coal gasification to meet future energy consumption. (Sutrisna, I.P., 2007). The research that I will take focuses on the syngas gasification results from the mine depth coal downdraft method, types of coal seams (A1, A2, B and C), and the effect of air flow rate.
1.1. Characteristics of Coal

Each type of coal has a different composition. Factors that cause this include the initial decomposition of coal origin through diagenetic processes and coalification [3]. The heterogeneity of coal properties is estimated in various relationships of its constituent components, such as the relationship between hydrogen and carbon content, oxygen and carbon content, and volatile content and calorific value. The quality of each coal deposit is determined by temperature and pressure and the length of time it forms, which is referred to as 'organic maturity'. Proximate coal content testing is needed to determine the character and composition of coal. Coal proximate analysis can be seen in Table 1:

Table 1. Proximate Analisys from each coal sample

| Type of Analisys | Type of Coal Seam |
|------------------|-------------------|
|                  | Tipe A1 | 6.57   | 6.63 | 2.09 | 2.10 |
|                  | Tipe A2 | 42.9   | 42.4 | 35.20 | 34.99 |
|                  | Tipe B  | 4.77   | 4.78 | 8.57 | 8.58 |
|                  | Tipe C  | 45.7   | 46.1 | 54.14 | 54.34 |

Table 2. Caloric Value of Coal Sample

| Type of Coal Seam | Caloric Value (Cal/gr) |
|-------------------|------------------------|
| Tipe A1           | 3481.6153              |
| Tipe A2           | 4541.4801              |
| Tipe B            | 4651.2393              |
| Tipe C            | 5217.9419              |

1.2. Coal Gasification

The thermochemical conversion of coal can be in the form of pyrolysis, gasification and combustion (combustion). The difference in the type of conversion lies in the amount of air (oxygen) consumed and the output during the conversion process. Gasification technology is a form of increasing the energy contained in coal through a conversion from a solid phase to a gas phase using a thermal degradation process of organic materials at high temperatures in incomplete combustion using limited air (20%-40% air), stoichiometry) [4].

The fuel used for the gasification process uses materials containing hydrocarbons such as coal and biomass. The entire gasification process occurs in the gasifier. In this gasifier, a heating process occurs to a certain reaction temperature and then the fuel goes through a combustion process by reacting with oxygen to produce combustible gas and other combustion products. Water vapor and carbon dioxide from combustion are reduced to flammable gases, namely carbon monoxide (CO), hydrogen (H2) and methane (CH4) which can then be used as power plants or stoves.

1.3. Downdraft Gasification

The development of gasification technology makes the process of research and development of gasifiers continue to be carried out. The development was carried out with various considerations, including reducing the tar and sulfur content in the syngas product. The downdraft gasifier is a reactor with the direction of the flow of air and raw materials both going down. Syngas flows down and gasifier. Stated that the reason for choosing the downdraft type gasifier was due to 4 things, namely:

1. Lower manufacturing costs,
2. The gas produced is hotter than the updraft system
3. Easier to proceed to the combustion process
4. Lower tar than updraft. This is because the tar which is the result of pyrolysis is carried along with the gas and then enters the combustion area (combustion) and then gasification which has a higher temperature. In this area of gasification and combustion, tar will then decompose.

1.4. Effect of Coal Rank on Syngas Products

The type of coal is very important to the desired syngas yield. Riza Abrar (2017) states that the conversion of H2 gas resulting from gasification is mostly produced.
by lignite type coal compared to subbituminous and anthracite coal types. However, for the type of coal that produces CO conversion, anthracite has the highest maximum CO conversion rate, followed by bituminous and lignite. This condition is based on the carbon content of the type of coal. Based on these levels, gasification is divided into 3 products, namely: Low-Btu gas (150-300 Btu/scf), Mediu-Btu gas (300-550 Btu/scf) and High-Btu gas (980-1080 Btu/scf). The product composition is shown in Table 3:

Table 3. Classification of Gasification Product

| Product                  | Composition                                                                 |
|--------------------------|------------------------------------------------------------------------------|
| Low-Btu gas (150-300 Btu/scf), | 50% ≥ nitrogen smallest component H2 dan CO, CO2 dan sedikit gas metana |
| Medium-Btu gas (300-550 Btu/scf) | Dominantly CO dan H2, and small unburner gas, smallest methane gas |
| High-Btu gas (980-1080 Btu/scf). | Pure all methane gas |

2. RESEARCH METHODS

2.1. Tools and Material

The tools and materials used in this research are a set of downdraft type coal gasification equipment, bomb calorimeter (Parr 6200), electric sieve shaker (Ziaulhaq Solution), furnace (Naberthem), gas analyzer, PTBA coal, and air.

2.2 Flow Diagram

![Flow diagram Of Gasification Process](image)

3. RESULT AND DISCUSSION

3.1. The Effect of Coal Layer Types on the Increase in Temperature and Flash Time

In Figure 3, it is shown that the highest temperature is type C coal seam, this is because type C coal has the highest heating value, which is 5217.9419 Cal/gr, and the water content in type C coal is low enough so that the temperature achieved higher than coal with a calorific value below 5217.9419 Cal/gr.

Pratama (2019) stated that the lower the water content of the raw materials and the higher the carbon value of the raw materials, the longer the flame of the syngas produced. Based on the research that has been done, the effect of the type of coal seam on the length of the flame is shown in Figure 4. there is an increase in the length of the flame from 2 minutes to 13 minutes. The shortest flame time is in type A1 coal with a heating value of 3481.6153 Cal/gr and the flame time is 2 minutes. While the longest flame time is in type C coal with a calorific value of 5217.9419 Cal/gr and a long flame time is 13 minutes.

![Graph Relationship Between Type Of coal With Flameable](image)
3.2. The Effect of Coal Types on Syngas Composition

Based on Figure 5, it can be seen that the high calorific value of the coal used can indicate the amount of carbon content in the coal. In his research shows that the value of the CH4 composition depends on the ultimate and proximate composition of the type of fuel used. Based on these factors, the higher the carbon value contained in a fuel, the more CH4 compounds will be formed. Meanwhile, the smaller the H2 value of the raw material causes the low H2 contained in the raw material. The high composition of oxygen in syngas is caused by the large diameter of the circle of the air intake pipe used, causing a lot of air to enter the combustion chamber.

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3.3. The Effect of Coal Types on Heat Heating Value Syngas

From the results of observations and tests that have been carried out, the following conclusions are obtained:

1. The fastest increase in temperature is shown by Type C coal with a heating value of 5217.9419 Cal/gr with a flame time of 13 minutes. The increase in temperature and duration of flame is caused by the low value of moisture contained in the coal.

2. The type of coal seam is one of the factors that affect the syngas produced. This factor is caused by the composition in the coal that affects the high and low value of CH4 gas that will be formed. Type C coal seam is a type of seam.

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