Research Article

Polyphenol Content and Antioxidant Activity of Stevia and Peppermint as a Result of Organic and Conventional Fertilization

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Stevia rebaudiana Bertoni and Mentha piperita are plants that generate interest mainly due to the presence of bioactive compounds in their leaves, such as phenolics. Studies indicate that phenolics have pharmacological and therapeutic properties, including antioxidant activity. Phenolic compounds may be affected by the type of fertilization. For this reason, organic and chemical fertilization were evaluated along with antioxidant activity. Results showed significant differences for total phenols in organic peppermint (62% higher content). Also, DPPH test displayed differences for peppermint and stevia (572% and 16% greater in organic). Organic fertilization may be alternative for producing high added agricultural and commercial products.

1. Introduction

Dietetics products and natural food ingredient demand has increased over the last years. A lot of interest has emerged on sources of natural antioxidant since many health problems are associated with the action of toxic forms of oxygen responsible for oxidation processes. Antioxidants are capable of inhibiting reactive oxygen species (ROS) [1]. Phenolic compounds may represent approximately 19–23% of dry peppermint leave weight [2]. It has been reported that phenolics show beneficial health effects due to free radical scavenging properties. Mexican population consumes infusions in a regular manner, and one of the most popular is prepared from peppermint (Mentha piperita). Peppermint leaves contain a wide array of bioactive components (fatty acids, volatile compounds, carotenoids, and phenolic compounds). At the same time, an interest in natural sweeteners has emerged due to the increasing health consciousness and concern related to sugar consumption and the problems related to the safety of some artificial non-nutritive sweeteners [3]. Stevia is an herb of Asteraceae family, which grows wild in South America, such as Paraguay and Brazil. Leaves are the economic part of the plant with a high concentration of steviol glycosides [4]. Stevia rebaudiana Bertoni has been used as a natural sweeter; it is categorized by high concentration of steviol glycosides in its leaves, which are up to 200 to 400 times sweeter than sucrose. In addition to their sweeteners, stevia plants possess other compounds such as terpenes, sterols, volatile acids, vitamins, carotenes, organic acids, polysaccharides, hormones, microelements, and phenolic compounds (tannins and flavonoids) [5].

Particularly important for the antioxidant capacity in stevia and peppermint are phenolic compounds, which are
secondary metabolites. The promotion of product quality regarding secondary metabolites content may be of high relevance for a commercial expansion of stevia and peppermint. Phenolic compounds are involved in various plant processes such as growth and reproduction and are also synthesized as a defense mechanism to various stresses; therefore, their production can be enhanced by different conditions, among them, type of fertilization. Nowadays, consumers are more concerned of possible exposure to agrochemicals. Organic agricultural practices do not allow the use of chemical compounds for crop nutrition, synthetic compounds for pest, disease, and weed control. Many have suggested that the use of organic fertilizers is an essential source of nutrients for sustainable agriculture and in addition to cover the physiological requirements of crops, favoring the development of high-quality crops [6]. Due to a plethora of implications, among them, health benefits and the use of environmental friendly agriculture practices [7], the aim of this study was investigating the effect of organic and chemical (conventional) fertilization on the content of bioactive compounds in *S. rebaudiana* and *M. piperita*. The work is focused on the analysis of antioxidant capacity, phenolic compounds, and steviosides levels in those materials, to assess differences between both types of fertilization.

2. Materials and Methods

2.1. Plant Material. Conventional peppermint and stevia plants were obtained from a local supplier. The plants belonged to the same batch. Organic stevia plants were got from a commercial greenhouse with organic care located at San José Iturbide, Guanajuato, México. Organic peppermint was grown at the Universidad Autónoma de Querétaro, Amazcalca campus. Leaves were collected and dried at 45°C for 24 h (Fisher Scientific, 650D, USA). Next, they were milled in a grinder (Krups GX4100, México).

2.2. Extract Preparation. Extraction for phenolics and antioxidant determinations were performed by placing 1 g (PRACTUM 224-1S; Sartorius, Göttingen, Germany) of fresh sample in a 50 mL tube and mixed with 10 mL of methanol. The tubes were protected from light and shaken at 200 rpm (Orbit 1000 model S2030-1000; Labnet, Woodbridge, NJ, USA) for 24 h at 25°C. After incubation, the samples were centrifuged (Sorvall Biofuge Primo R model 75005448; Thermo Scientific, Osterode, Germany) at 6,793 × g for 10 min. Aliquots of the supernatant were taken for the assays. We followed the methods of Garcia-Mier, Jimenez-Garcia [8].

2.3. Quantification of Condensed Tannins. Condensed tannins expressed as milligrams of (+)-catechin equivalents per gram of dry sample were quantified according to the next procedure proposed by Garcia-Mier, Jimenez-Garcia [8]. Briefly, 200 μL of vanillin reagent (1% vanillin, 8% HCl in methanol) was added to 50 μL of methanolic extract and placed in a 96-well plate; each sample was tested in triplicate. Condensed tannins were quantified at 492 nm in a microplate reader (Multiskan Go model 51119300; Thermo Scientific, Vantaa, Finland) using (+)-catechin (up to 0.1 mg·mL⁻¹) as a reference standard. A blank sample was prepared by subjecting the original extract to the same conditions of reaction without the vanillin reagent.

2.4. Quantification of Flavonoids. Briefly, the method for the determination of flavonoids content was performed according to Garcia-Mier, Jimenez-Garcia [8]. It consisted of mixing 50 μL of the methanolic extract with 180 μL of distilled water and 20 μL of a solution 2-amino-ethylidiphenylborate 1% in a 96-well plate. The absorbance of the solution was monitored at 404 nm with a microplate reader (Multiskan Go model 51119300; Thermo Scientific, Vantaa, Finland). A rutin standard was prepared in methanol. Extract absorption was compared with that of a rutin standard curve (up to 2 μg·mL⁻¹). Flavonoid content was expressed as mg of rutin equivalent per gram of dry sample.

2.5. Quantification of Total Phenols. Total phenols (expressed as mg of gallic acid equivalent per gram of dry sample) were determined by the Folin-Ciocalteu method with modifications. To 40 μL of extract were added 460 μL of distilled water, 250 μL of Folin Ciocalteu reagent, and 1250 μL of 20% sodium carbonate solution. After 2 hours in the dark, samples were read at 750 nm in a UV/Vis spectrophotometer (Genesys 10S UV-Vis, Thermo Fischer Scientific, USA). Gallic acid was used for the calibration curve [9].

2.6. Antioxidant Activity by 1,1-Diphenyl-2-picrylhydrazyl Radical (DPPH) Inhibition Assay. Radical scavenging activity (RSA) was determined using stable radical DPPH method. The assay was performed following the next procedure. All reactions were conducted in 96-well microplates. Aliquot (20 μL) of methanolic extracts was mixed with 100 μM of DPPH (200 μL) in methanol. It was used a control and a blank. After 30-minute incubation at ambient temperature in darkness, absorbance was recorded at 515 nm in a microplate reader (Multiskan Go model 51119300; Thermo Scientific, Vantaa, Finland). It was prepared as a calibration curve with Trolox. The antioxidant activity was expressed as percent of inhibition [8].

2.7. Antioxidant Activity by 2,2′-Azino-bis(3-ethyl-benzothiazoline-6-sulphonic Acid) (ABTS) Inhibition Assay. The radical cation was prepared by mixing 7 mM ABTS stock solution with 140 mM potassium persulfate (1/1, v/v) and leaving the mixture for 12 h until reaction was completed and the absorbance was stable. The ABTS' solution was diluted with ethanol to an absorbance of 0.700 ± 0.05 nm at 730 nm for measurement. The photometric assay was conducted on 0.9 mL of ABTS solution and 0.1 mL of extract and mixed for 45 seconds, and measurements were taken at 730 nm after 15 minutes. The antioxidant activity of the sample was calculated by determining the decrease in absorbance. Trolox was used as standard substance. This assay was based on the ability of different substances to scavenge...
radicals. The antioxidant activity was expressed as percent of inhibition [10].

2.8. HPLC Analysis. For extraction, purification, and quantification of stevia extracts, the methodology used was proposed by Mondal, Majumdar [11]. It consisted of S. rebaudiana freeze-dried leaves (1 g) that were mixed with 10 mL of a mobile phase (acetonitrile: water 80:20) for 20 minutes. Rebaudioside A was identified and quantified by the high-pressure liquid chromatography method (HPLC, Hewlett Packard 1100 model), in which 20 μL was taken and was injected into the chromatography equipment. The used column was Zorbax Carbohydrate with a flow of 0.1 mL/min.

2.9. Statistical Analysis. Data were subjected to analysis of variance (ANOVA) followed by Student’s t-test (with P ≤ 0.05) by JMP (SAS Institute Inc., Cary, NC, USA).

3. Results and Discussion

3.1. Phenolic Compounds. Phenolic compounds, flavonoids, and condensed tannins as well as antioxidant activity were analyzed in stevia and peppermint by spectrophotometric methods after the application of organic and conventional fertilization. A number of reports indicate that these compounds are involved in the prevention of non-transmissible chronic diseases by means of their antioxidant activity [12, 13]. The total phenol, flavonoid, and tannin content for stevia and peppermint are shown in Tables 1 and 2. Results indicate significant differences between organic and conventional fertilization in peppermint for total phenolics but not for flavonoids and condensed tannins; these variables do not display significant differences in stevia; nevertheless, it is relevant to mention that organic stevia has 33% more total phenols than the conventional one. In the high total phenolic content in peppermint, other phenolics may be implicated apart from flavonoids and condensed tannins such as phenolic acids, and caffeic acid derivatives were the most abundant phenolic compounds in stevia [14]; also, caffeic acid was the most abundant phenolic acid in peppermint [15]; hydroxycinnamate derivatives have been identified and quantified in stevia methanolic extracts [2, 16]. Various authors have stated that the use of organic fertilizer enhances the amount of secondary metabolites such as phenolic as well the antioxidant activity in crops [17, 18]. The amount of stevia phenolic compounds found in this study is considerably lower than those reported by other studies (e.g., total phenol: 80.13 mg gallic acid/g extract and total flavonoids: 111.16 mg quercetin/g extract) [5, 19, 20]. The same behavior occurs for peppermint, where total phenolic and total flavonoid in peppermint leaves were 360.04 ± 0.285 and 421.96 ± 0.25 mg 100 g⁻¹, respectively [21]. According to Gupta et al. [22], tannin detected content in stevia leaves is 5.68 mg 100 g⁻¹ what is higher than the amount found in this work. Sujana et al. [23] reported 2 mg·g⁻¹ of tannins; this value is higher than the one found in this research. Higher levels of total phenolic content were consistently found in organic and sustainable marionberries, strawberries, and corn as compared to those produced by conventional agricultural practices [24]. According to the study of Faller and Fialho [25] made on fruits and vegetables, organic agriculture results in food products with similar or marginally higher polyphenol content and antioxidant capacity. The perception among consumers is that organic cultivars possess a higher nutritional quality than conventional; nevertheless, it is not easy to estimate compositional differences due to agricultural practices because of the vast number of variables such as crop, irrigation patterns, weather variations, handling, etc. Post-harvest management and also laboratory extraction techniques may be implicated in these differences [6].

3.2. Antioxidant Activity. Antioxidant capacity was determined in extracts from leaves of S. rebaudiana Bertoni and M. piperita. This activity was characterized by two different assays, consisting of measuring the ability to scavenge the DPPH radical. Results obtained are shown in Tables 3 and 4 for stevia and peppermint, respectively. A different behavior between these two tests can be observed. In the ABTS test, no significant differences were observed. However, DPPH test displayed significant differences for both plants. Stevia presented 0.880 ± 0.009 and 0.827 ± 0.16 mg Trolox equivalent g of extract⁻¹ for organic and chemical fertilization, respectively. Peppermint has 0.889 ± 0.014 and 0.558 ± 0.028 mg Trolox equivalent g of extract⁻¹ for organic and conventional fertilization, respectively. These results are lower than those presented in other studies [13]. This may be related to the lower amount of phenolic compounds founds in the material tested. Yildiz-Ozturk et al. [19] found DPPH radical scavenging activities of around 90% which are higher than the ones found; nevertheless, Singh et al. [26] reported radical scavenging activities (%) that ranged between 47.1% and 82.4% for leaf and flower, respectively. Both assays are classified into the group of single electron transfer based methods, which accounts for the similarity of results reported in many occasions. However, it must be considered that the conditions in which the tests are performed are different, so they must be considered complementary [27]. Total phenolic content correlates positively with antioxidant activity. Other studies have found the same relation. It is stated that antioxidant capacity characterizing stevia leaves is mainly attributable to the presence phenolic compounds, such as flavonoids [2] and steviosides [28]; Barroso et al. [14] stated the total flavonoid content was less strongly correlated with the antioxidant activity in comparison with the total phenolic acids, due to its higher concentration present in stevia. The phenolic compounds present in the herbs and spices have been reported to show natural antioxidant activity and are applied as food preservatives. The predominant mode of antioxidant activity of phenolic compounds is believed to be a radical scavenging via hydrogen donation [5].

Organic fertilization may enhance some bioactive compounds that correlate with antioxidant activity in plants; however, more research is needed to ensure getting high
added value agricultural products and nutraceutical products. In some cases, the strategy is not just organic or conventional but integrative, as the one exposed for peppermint [29]. More foods must be evaluated to solve the controversy of whether or not the strategy is not just organic or conventional but integrative.

Due to their amount of steviol glycosides, which are of great medicinal and nutraceutical importance worldwide, two different varieties of stevia (Morita and Eriete) were evaluated in their amount of glycosides (stevioside, rebaudioside A, and rebaudioside C) under organic and conventional fertilization. For organic fertilization, in Morita, steviol glycoside content expressed as mg/g of dry matter was 9.582, 1.184, and 7.496, respectively. For Eriete, it was 1.160, 24.719, and 5.748, respectively. For Morita cultivated under conventional fertilization, steviol glycosides were not detected and rebaudioside A was found in an amount of 14.877 mg/g of dry matter and rebaudioside C in 5.893. Eriete under this same condition presented 25.710 mg/g of dry matter for rebaudioside A and 4.891 mg/g of dry matter for rebaudioside C. No differences in the steviol glycoside content were found between varieties or fertilization. Diaz-Gutierrez et al. [30] found concentration of rebaudioside C and stevioside increased under organo-mineral fertilizer (compost from poultry manure with inorganic salts) under greenhouse conditions. These authors found that nutrients as nitrogen, calcium, magnesium, and sulfur presented a significant correlation with the production of steviol glycosides. The implication of nitrogen fertilization in the production of steviol glycosides and other stevia bioactives has been also stated by other authors [14, 28], implying that adequate N rate is important to significantly increase and optimize the bioactive compound levels in stevia; however, it was stated by Barbet-Massin et al. [31] that steviol concentration in the leaf decreased with increasing nitrogen concentration; then, the implication of nitrogen in the development of steviol glycosides needs further evaluation since it also implies that an N deficit could be switching between the production of steviol glycoside and the synthesis of key isoprenoids, such as gibberellins, chlorophylls, and carotenoids.

It is identified that, among steviol glycosides, stevioside and rebaudioside A are present in the highest concentrations. Stevioside is 300 times sweeter than sucrose; it has a slight licorice flavor and a bitter aftertaste. In contrast, rebaudioside A lacks the bitter aftertaste and is 250 up to 400 times sweeter than sucrose [31]; for this reason, the rebaudioside A/stevioside A ratio is considered a parameter to measure the quality of stevia extract [32]. For Morita cultivated under organic condition, rebaudioside A/stevioside A ratio was of 0.12, whereas for Eriete it was 21.31. In cultivars grown in conventional manner, no stevioside was detected. The results for Eriete are relevant since it will develop a better taste in steviol glycoside products. It is relevant to mention that the ratio reported here is higher than the one reported in a study where mannitol, an NaCl stress, was considered. A significant difference of 0.05 by Student’s t-test.
and peppermint in order to provide better guidance to agricultures and consumers.

Data Availability

The data are available from the corresponding author upon request.

Conflicts of Interest

The authors declare that there are no conflicts of interest regarding the publication of this paper.

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