A phytomass-inspired carbon and its importance as an antibacterial agent against human pathogens

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Abstract

Background: Imprudent use of antimicrobial drugs has resulted in the microbial resistance among the known microbes and hence we foresee a pressing need towards the development of novel, low-cost, and high potent antimicrobials which should be munificent by nature. In the pursuit of the above, phosphoric acid activated low-cost carbon was produced from a renewable phytomass precursor viz., leaves of Vitex negundo L. plant and explored for its antibacterial efficacy against four human pathogens viz., S. aureus, S. pyogenes (Gram-positive bacteria), and E. coli, P. aeruginosa (Gram-negative bacteria) by adopting well diffusion method. Carbon yield, burn-off, phase purity, elemental composition, particle morphology, and surface functionalities have been studied by ultimate elemental analysis, X-ray diffractometry, elemental analysis, scanning electron microscopy, and Fourier transform infrared spectrophotometry respectively. Minimal inhibition concentration (MIC) was also followed. Plausible mechanism of killing the pathogens by the above activated carbon was also provided.

Results: Vitex negundo leaves derived activated carbon (VNLAC) was found to contain large number of O-, S- and N-containing surface groups which are supposedly responsible for bestowing antibacterial properties to the carbon derived from Vitex negundo leaves. It has emerged as a potential antibacterial agent for many Gram-negative as well as Gram-positive bacteria. The inhibition zone of mean diameters ranged from 9 to 25 mm against all the pathogens was significantly (p < 0.05) less than that of the control viz., ciprofloxacin. Thus, the fundamental experimental results may extend the limits of carbon sources but also the conventional idea of obtaining active carbon to apply in technologies where carbon is inevitable.

Conclusion: The work not only demonstrates the promising potential of VNLAC as an efficient antibacterial agent but also presents a feasible mechanism of action of removing pathogens. Vitex negundo-derived carbon may become a cheap substitute for cost-prohibitive drugs. The findings of the work illustrate an easy choice as an antibacterial for topical application at infected sites.

Keywords: Phytomass, Activated carbon, Pathogens, Antibacterial efficacy, Vitex negundo L

Background

It is well known that antimicrobials may act like killing agents or impede the growth of microorganisms. On the one hand, due to the extraordinary recombination capability of the microorganisms, the genetic materials undergo rapid mutagenic transformations that hinder metabolic processes such as cell wall formation, synthesis and transcription of proteins, thus leading to the development of resistance against the known antimicrobials or antibacterials in the case of bacteria. On the other hand, recent unwise use of antibiotics has also resulted in the multitude of pathogens to develop resistance to the commonly employed antimicrobials. Therefore, pollution due to microbes is becoming one of the major threats related to environmental sanitation.
and human health ultimately. As a consequence, there arises a compelling and dire need for the research and development of novel and high potential antimicrobials which should be benevolent by nature.

Now turning to the alternatives available for the above subject, recent literature has shown that nanoparticles based on carbon nanotubes, fullerenes, and graphene oxide possess appreciable antimicrobial activity against various strains [1]. Of late, antimicrobial activities of carbon-based nanostructures are being interestingly investigated owing to their high surface to volume ratio, large inner volume, and for other unique physical and chemical properties. Therefore, technological developments towards the design and synthesis of carbon-based antimicrobials that too from waste biomass that will reduce microbial resistance and infection burden have zeroed-in recently. Falling in this line, the present work explores the prospects of utilizing zero-cost leaf biomass of Vitex negundo L. plant for deriving activated carbon for evaluating its antibacterial properties.

Generally decayed or decaying plant biomass poses eco-pollution as its water content, sugar, mineral, and proteins would favor exponential growth of microorganisms. Also, a lot of microorganisms breed before the biomass could get decomposed, eventually bringing out rotten smell thereby polluting the environment. Generally, decayed or decaying plant biomass pose environmental pollution as its sugar, mineral, protein contents, and water retention nature creates favorable conditions for proliferation of microorganisms.

Phytochemical constituents of medicinal plants have been considered to be a basic requirement in the discovery of potential medicines and remedies for various diseases in Ayurvedic and Nutraceutical research. Medicinal properties of any plant depend on the type of phyto constituents and nutritive elements as well as minerals. Medicinal plants are effective against various health problems due to their pharmacological efficacy which depends on their elemental concentrations [2]. Speaking about Vitex negundo, only medicinal properties are more frequently researched. The whole plant is traditionally important because it has many therapeutic values. It has been reported to possess potential pharmacological properties like antiinflammatory, antirheumatic, and antibiotic, hepato protective, antioxidant, anticonvulsant, oxidative stress, snake venom neutralization, and antiallergic activities [3]. The leaves of Vitex Negundo is used to treat sprain, headache, abdominal pain, tooth ache, asthma, cough, fever, boils, wounds, and ulcers due to its antiseptic, astringent, anti-inflammatory, and antipyretic nature [4]. Thus, it can be seen that Vitex negundo plant has intense medicinal values and literature is also enormous. Hitherto, research is being focused only on the phytochemical composition and medicinal values of the plant Vitex negundo. However, one or two reports on activated carbon derived from Vitex negundo bark, stem, and flowers are only available [5].

Thus, it is apparent that the phytomass (plant biomass) wastes which may not have economic benefits may wisely be utilized by transforming zero-cost phytomass into activated carbon, which doubles up its worth by minimizing the waste management cost and by appreciably providing inexpensive alternative to the existing high priced commercial functional carbons. Thus, plant wastes from multiple sources present cost free and renewable carbon resource for numerous applications such as in biomedical and pharmaceutical fields as antibacterial agent.

The study of natural compounds containing sulphur and nitrogen is interesting because of their significant antibacterial, antifungal, and anticancer activity [6]. It can be shown later in this article that the biomass-derived activated carbon too has sulphur and nitrogen (in addition to carbon, hydrogen, and oxygen) which are present as organic functional moieties in its structure to explain the observed effect against the pathogens. During the past 5 years, not many articles have appeared in the literature describing the biological activities using biomass-derived activated carbon alone. But to cite a few examples, using groundnut shell, Yallappa et al. [7] have done a pioneering work in the regard above. Lakshmi et al. have elaborately reviewed activated carbon nanoparticles from biowaste as a new generation antimicrobial agents [8]. It is interesting to know that the kitchen soot was once applied as an antimicrobial and called “old woman’s remedy” [9]. In that sense, Sheena et al. [9] have reported antimicrobial activity of carbon nanoparticles isolated from natural sources against pathogenic Gram-negative and Gram-positive bacteria. Sekaran et al. [10] have reported the preparation of mesoporous-activated carbon from rice husk by precarbonization at 400 °C, chemical activation using phosphoric acid at various temperatures and have immobilized Bacillus sp. in the mesoporous-activated carbon for the degradation of sulphonated phenolic compounds in wastewater. Karthik et al. [11] have prepared activated carbon from Tribulus terrestris and have shown activity against E. coli, B. subtilis, S. aureus, and K. pneumoniae. Shamsi et al. [12] have reported a clear zone of inhibition of carbon nanoparticles obtained from sandal wood bark against B. cereus, E. coli, C. violeceum, and P. notatum due to these nanoparticles. Dheepan et al. [13] have reported in vitro evaluation of antibacterial efficacy using passiflora foetida-activated carbon against a score of pathogens. All these reports conclude that carbon materials produced from biomass still has enormous potential for acting against various strains of
bacteria and that extended research is needed for a commercial success in the near future. Therefore, it is definite that the research outcome will be a welcoming route for producing zero-cost phytomass-derived cheap and affordable carbon materials possessing favorable characteristics for the intended applications.

The present research paper accordingly covers the preparation of activated carbon from natural and green source namely the leaves of *V. negundo* plant, the zero-cost phytomass and the evaluation of antimicrobial activity against selected human bacterial pathogens like *S. aureus*, *S. pyogenes* (Gram-positive type), and *E. coli*, *P. aeruginosa* (Gram-negative type). It further provides an ideal method for the utilization of *V. negundo* plant for producing low-cost and multifunctional-activated carbon which can be produced in bulk. This work is an evidence of taking wastes-making products-turning them into resources (circular economy), the opposite of the current setup of seizing resources-making products-turning them to wastes (linear economy).

**Methods**

The raw material used for preparing biocarbon was the leaf biomass of *Vitex negundo* L. (*V. negundo*) plant which is obviously economical, abundant and easily available. Pathogens selected for the present investigation are *S. aureus*, *S. pyogenes* (Gram-positive bacteria) and *E. coli*, *P. aeruginosa* (Gram-negative bacteria).

**Collection of V. negundo leaf biomass**

The biomass was collected from a farm land in Batlagundu, Dindigul District, Tamil Nadu, India. A sample of the leaf bearing voucher Specimen No. NA/MADU/CHEMDDEMKU/021/07/2018-19 was deposited and preserved at the Department of Botany of one of the reputed Institutes, Tamil Nadu, India, and was identified and authenticated by a botanist/taxonomist of the same institute. The biomass was cleaned well in double distilled water, drained, and dried under shade for a week. Later, the dried leaves were powdered using a domestic blender to prepare activated carbon. The photograph of biomass precursor is given in Fig. 1.

Reagents and chemicals used in the investigation are of analytical grade and purchased from Merck Ltd., India. Doubly distilled water was produced from our laboratory.

**Preparation of V. negundo leaves activated carbon (VNLC)-carbonization (activation) with H₃PO₄**

In a typical preparation, 30 g of the dried biomass powder was impregnated with H₃PO₄ solution in water (in the ratio 1:1 (w/w)) by heating the powder-acid mixture at 60 °C for 48 h and kept aside for 24 h. The dried lump was pyrolyzed at 350 °C for 2 h in a muffle furnace, cooled, powdered using mortar, and finally washed with plenty of hot doubly distilled water to remove the residual acid until the washings are pH neutral and its conductivity is minimal. pH neutrality and low conductivity values of the washings ensure thorough washing of the sample. The mixture is filtered and the resultant black material is thoroughly dried, powdered, and sifted to 250 mesh size. The fine powder is now designated as VNLC. The sample was stored in an air proof vial for further use. The preparation methodology of VNLC has been presented in the flow chart (Fig. 2).

**Physio-chemical characterization studies**

Physio-chemical characterization studies which include burn-off, elemental analysis, powder X-ray diffractometry (PXRD), Fourier transform infrared spectroscopy (FTIR), and scanning electron microscopy (SEM) were performed on VNLC sample. Elemental analysis was done on Vario ELIII CHNS/O elemental analyzer. X-ray diffraction pattern of VNLC was recorded on XPert Pro X-ray diffractometer with CuKα radiation source between 10 and 80°. The morphology of VNLC was examined with Hitachi S-4700 model scanning electron microscope (SEM). Surfacial functional organic moieties on the surface of the carbon were ascertained with FTIR spectrometer Model # Nexus 670 from frequency 4000 to 400 cm⁻¹.

**Determination of minimal inhibition concentration (MIC)**

Minimal inhibition concentration (MIC) is considered as the standard procedure for determining the susceptibility of microorganisms to an antimicrobial. MIC is defined as the lowest concentration of an antibiotic (antibacterial in this study) that will inhibit the visible growth of bacteria after overnight incubation [14]. For the determination of MIC, the method of dilution was followed.
typical procedure, an extract was prepared by dispersing 50 to 90 μg of VNLAC in 1 mL of 5% DMSO-water mixture with sterile nutrient broth in sterilized glass test tubes by sonicating for 30 min. By means of a standard and sterilized wire loop, purchased from Merck, India, a loop full (10 μL) of E. coli culture, for example, was inoculated into the test tubes containing 1 mL of the various concentrations of the VNLAC mixture in the nutrient broth. Similar procedure was followed for the rest of the strains. The test tubes were incubated at 37 °C for 20 to 24 h and observed for growth or turbidity through unaided eye.

Data analysis
The zone of inhibition data are presented as a mean of the triplicate measurements ± standard error of mean (SEM). Statistical Package for Social Sciences (SPSS version 20.0) software was used for the data analysis. The statistical difference of the mean zone of inhibition due to the four pathogens was independently carried out by one-way analysis of variance (ANOVA) at a significant level of p < 0.05.
Results
Activated carbon sample produced from phosphoric acid activation of the leaf biomass of *V. negundo* plant was evaluated by diverse techniques and the noteworthy results are presented in the following sections.

Physico-chemical features of VNLAC

Yield and burn-off
The % yield of the activated carbon is the ratio of the weight of the activated carbon (*w*) obtained to that of the dry and powdered biomass (*w₀*) taken and is given by the following formula:

\[
\text{Yield} \% = \left( \frac{w}{w₀} \right) \times 100\%.
\]

The burn-off is the weight loss percentage due to the activation step. The yield of VNLAC was calculated as 66% and burn-off as 34%.

Ultimate elemental (CHNS) analysis
The VNLAC was subjected to elemental analysis and observed to contain 60.49% C, 9.36% N, 0.29% S, and 6.98% H. Earlier reports [15, 16] about thermally treated carbon samples the remaining element in the carbon material should be O. This result has been presented in Fig. 3.

Phase analysis by powder X-ray diffractometry (PXRD)
XRD pattern of VNLAC is shown in Fig. 4. The broad reflection between 23° and 30° indicates (002) diffraction peak, ascribed to the amorphous and low graphitization nature of VNLAC. In addition, the peak indicates the development of microporous carbon, which is mostly amorphous also with non-crystalline features to a higher degree [17]. It is presumed that a large number of disordered single graphene layers and stacked structures of graphene sheets may be present simultaneously in the texture of the carbon powder [18], and the slightly broad shape also indicates the highly disordered structure in the carbon [19].

The evolution of graphitic structure in VNLAC may be described by the fact that the thermal decomposition of various organic compounds in the biomass results in cleavage of chemical bonds and repolymerization of radicals, further condensing into carbon compounds having characteristic graphitic layers and stacked sheets of graphene [20]. As of now, the authors reserve the study on the correlation between biological activity and the extent of graphitization in the VNLAC.

Morphology by scanning electron microscopy (SEM)
Scanning electron microscopy (SEM) was employed to identify the particle morphology of the VNLAC powder and Fig. 5 depicts the same.

SEM of VNLAC sample infers that the particles are nearly spherical in shape with lumps distributed sparsely yet compacted together yielding a fine network of all particles. Porosity is also obvious from the SEM. H₃PO₄ might have acted aggressively on the cellular features of the precursor biomass resulting in enhanced carbon gasification. It is known that the pores in the chemically activated carbon samples may be reasoned to the paths of the escaping gases (which leave spaces during carbonization) as well as the subsequent removal of unreacted activating chemical by water washing. It has to be stressed that morphology of any biomass-derived carbon samples depends on the activation conditions.

Fourier transform infrared (FTIR) vibrational studies
In addition to various physical features studied above, the activity of the carbon sample depends also on the chemical structure of the functional groups present at the carbon surface. Hence, knowledge on the surface functional groups gives an insight to the functionality and also the reactivity of the activated carbon. So FTIR data was collected for qualitative chemical information on the surface organic functional groups that are present in VNLAC sample and has been presented in Fig. 6 and data in Table 1.

It is well documented that plant contains phytochemicals like flavonoids, tannins, saponin, glycosides, terpenoids, steroids, quinines, amino acids, and proteins which are responsible for certain physio-chemico as well as biological properties. But during carbonization or pyrolysis of the biomass, these chemicals would tend to decompose and gets converted into various organic functional moieties which decorate the surface of the AC as evidenced from FTIR studies. As a result, FTIR spectrum of VNLAC is also complex due to the presence of many organic functional groups on the carbon surface. The spectrum obtained for VNLAC shows a characteristic signature around 3700 cm⁻¹ which is endorsed
to free O–H groups and the peak observed around 3500–3300 cm\(^{-1}\) is due to the NH stretching of amino group [21]. A peak around 2260–2220 cm\(^{-1}\) might be due to the –C≡N stretching [21] and the IR signature at 1720 cm\(^{-1}\) could be assigned to C=O stretching of carboxyl groups [22]. The peaks around 1650–1580 cm\(^{-1}\) are due to the NH bending or O–H bending vibrations of absorbed water with some carboxyl groups [23]. The carboxyl groups are presumed to provide excellent dispersivity [24]. Peak at 1627 cm\(^{-1}\) could be due to the C=C ring skeletal stretching of aromatic carbons [25] and IR signature at 1460 cm\(^{-1}\) could be assigned to C–H in plane bending vibration of allylic carbon [26]. The band around 1370 cm\(^{-1}\) is due to C–H stretching vibration of gem dimethyl group. The band evidenced around 1360 cm\(^{-1}\) is for S=O in sulphoxide group [27]. Peak at 1150 cm\(^{-1}\) shows C–O stretching vibration of alcohol and 1260, 1210 cm\(^{-1}\) represents the C–O–C stretching vibration of ether [28]. A band registered from 700 to 600 cm\(^{-1}\) has been due to C–H out of plane bending [22].

Thus O, H, N, and S elements do exist on the surface of the VNLAC as amines, sulphoxide, carboxyls, amino, and alcholic functional groups. These functional groups present in the carbon matrix of VNLAC are presumed to be bonded to the edges of the carbon layers and may govern the surface chemistry of the carbon [29, 30], which in turn may have a say on the bactericidal properties and efficacy. Figure 7 represents a schematic view of the various organic functional groups that are presumed to be on the surface of the activated carbon.

Results of minimal inhibition concentration (MIC)
It is to be noted that the test tubes containing concentrations of antibiotic or antibacterial that showed no bacterial growth or turbidity after the first 24 h of incubation showed turbidity after the addition of equal volumes of sterile nutrient broth and further 24 h of incubation were said to have minimum inhibitory concentration.

Results of antimicrobial activity of VNLAC against pathogens
The antimicrobial efficacy were studied in different loading levels of VNLAC (100, 200, 300, 400, and 500 μg/
mL) in 5% DMSO against four pathogenic bacterial strains viz., two Gram-positive (S. aureus, S. pyogenes) and two Gram-negative (E. coli, P. aeruginosa). Antimicrobial activity of VNLAC was studied using agar well diffusion method. The study was performed by sterilizing Mueller-Hinton agar media (from Himedia) Petri plates. After solidification, 6 mm wells were cut on the Mueller-Hinton agar using a sterilized cork borer. The two clinical test bacterial pathogens were separately swabbed onto the surface of the agar petri plates. Wells were impregnated with 20 μL of the test samples with different loading from 100 to 500 μg of sample/mL of 5% DMSO. The plates were then incubated at ambient temperature for 30 min to allow the compound to diffuse into the medium. Finally, the plates were incubated or developed at 30 °C for 24 h to get the antibacterial activity evaluated by measuring the zone of inhibition around the wells.

Discussion

Yield and burn-off
It is a common understanding that the % yield of the activated carbon from the raw material would decrease with increase in temperature and so the % burn-off would tend to increase with increase in temperature. It is worth mentioning that the heat processing of the biomass was carried out in a muffle furnace where there is a free access to air or oxygen; significant amount of carbon content might have been converted as CO₂. Nevertheless, increased yields could have been obtained if the sample was heated in an inert atmosphere.

Ultimate elemental (CHNS) analysis
The result of the elemental analysis has been presented in Fig. 3. The presence of N, S, and O shows the existence of various organic functional groups on the surface of the VNLAC sample. The presence of several organic groups containing the hetero elements is also corroborated

![SEM image of VNLAC](image1.png)

![FTIR spectra of VNLAC](image2.png)
through vibration spectroscopic studies (FTIR) and is expected to contribute to the overall performance of the VNLAC for the applications for which it is proposed for. The organic groups are rigorously considered in explaining the mechanism of biocidal activity in addition to physical features of the VNLAC powder.

**Minimal inhibition concentration**
From Table 2, it can be inferred that all bacterial strains showed turbidity after incubation in nutrient broth with different VNLAC concentrations ranging from 50 through 80 μg/mL. After 80 μg/mL, no turbidity was observed for all bacterial strains and indicates no further bacterial growth. Hence, 80 μg/mL has been considered as the MIC for all the bacterial strains.

**Antimicrobial activity of VNLAC against pathogens**
Each antibacterial assay was performed in triplicate and mean values, which represents the inhibitory effect of VNLAC±SEM is presented in Table 3. These observations are presented in Fig. 8a (Gram-positive pathogens) and Fig. 8b (Gram-negative pathogens) also.

The antibacterial activity of VNLAC increased linearly with the increase in loading of VNLAC (μg/mL). Compared with the standard ciprofloxacin, the results discovered better antibacterial activity of VNLAC against all the four pathogens studied but Gram-negative types were more active than Gram-positive types. It can be observed from Table 2 that the inhibition zone measured for all the bacteria ranged from 9 to 25 mm ± SEM. Also, the zone of inhibition of VNLAC at 500 μg/mL is not significantly different from the value at 400 or 300 μg/mL, but it is significantly different at 100 μg/mL against the tested bacteria at $p < 0.05$. All the tested bacteria in all concentrations of VNLAC were significantly different from that of the positive control namely Ciprofloxacin (standard 30 μg disc) at $p < 0.05$ significant level. As expected, the activity was found to be concentration

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### Table 1 FTIR signatures and assignment of FTIR signals of VNLAC

| IR frequency (cm$^{-1}$) | Assignment |
|--------------------------|------------|
| 610–700                  | –C–H bend, –C=S stretching of sulphoxides |
| 1150                     | –C–O stretching vibration of alcohol |
| 1260, 1211               | –C–O–C stretching vibration of ether |
| 1360                     | –S=O stretching of sulphoxides |
| 1370                     | –C–H stretching vibration of gem dimethyl group |
| 1460                     | –C–H in plane bending vibration of allylic group |
| 1627                     | C=C ring skeletal stretching vibration of aromatic structure |
| 1650–1580                | NH bending, O–H bending vibrations of absorbed water with some carboxyl groups |
| 1720                     | –C=O stretching vibration of carbonyl group |
| 2260–2220                | –C≡N stretching |
| 3500–3300                | –NH stretching of amino group |
| 3700                     | OH stretching |

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![Fig. 7 Model of various organic functional groups on the surface of the activated carbon.](image)
dependent, as the zone of inhibition increases as we move from 100 to 500 μg/mL, as evidenced from Table 3.

As discussed above, VNLAC is more active against Gram-negative types than Gram-positive types. The difference in the inhibitory action of VNLAC against the Gram-positive types studied may be due to the structure of the cell membrane of the Gram pathogens pathogens, where it may be difficult for the particles of VNLAC to move across or break the cell wall to disintegrate the intercellular components of the pathogenic cells. Obviously, the bacterial growth in the control plates (solvent only) was not hindered at all thereby confirming that VNLAC sample possesses better bactericidal activity.

Probable mechanism of action of activated carbon on microbes

- Microorganisms may adsorb/adhere to carbon surface through strong Van der Waals forces.
- Carbon particles may cross the cell membrane of the bacteria, penetrates into the cell, interacts with intracellular matters, weakens DNA replication thereby inactivating proteins, and prevents bacteria from undergoing cell division finally leading to cell lysis. As a consequence, bacteria cannot easily develop resistance [31].
- Cell-C aggregates formation may cause damage to the cell membrane of bacteria and then release of their DNA content [31].

Arias et al. [32] have showed that carbon with surface groups of –OH and –COOH indicated improved antimicrobial activity to both Gram-positive and Gram-negative bacteria. They have also confirmed that antimicrobial activity is carbon particles shape and size dependent. VNLAC sample is also observed to contain these groups and hence the observed antibacterial activity may be explained.

- Effect of sulphur- and nitrogen-containing groups on the antimicrobial activity leaves an opportunity for further research.

Figure 9 represents the probable broad mechanism of antibacterial activity of VNLAC envisaged from the facts discussed above.

One thrust area of study in combating bacterial infections and diseases is the development of affordable and munificent drugs and the choice of using the available drugs with structural modifications. Although some universal guidelines for the treatment of diseases caused by microbes are accepted everywhere, it seems necessary to search for new antimicrobial agents for reasons such as

1. Infecting microbes becoming resistant to present drugs
2. Some drugs producing side effects
3. The present treatment of immune depressed patients being grossly remains unsatisfactory

It is to be noted that recent knowledge on the mechanism of action of the drugs available and the biochemical mechanisms of resistance against them may be a useful basis for designing new and better drugs to fight bacterial diseases. The knowledge of the mechanism of microbial resistance to a drug could indicate the direction in which search for new products that specially overcome this mechanism. Thus, we hope that the information provided in this research paper proves valuable, stimulates broader concerns, and invigorates further developments in this promising field.

The present research suggests that there is no need of complex activation procedure for the production of phytomass-derived carbon with antibacterial activity.

Table 2 Inhibition of bacterial strains using different concentrations of VNLAC in broth after 24 h of incubation at 37 °C

| Bacterial strains tested | Turbidity in broth Conc. of VNLAC (μg/mL) | 50 | 60 | 70 | 80 | 90 |
|-------------------------|------------------------------------------|----|----|----|----|----|
| S. pyogenes             |                                          | +  | +  | +  | +  | −  |
| S. aureus               |                                          | +  | +  | +  | +  | −  |
| E. coli                 |                                          | +  | +  | +  | +  | −  |
| P. aeruginosa           |                                          | +  | +  | +  | +  | −  |

+ indicates growth; − indicates no growth

Table 3 Susceptibility of various pathogens to growth inhibition by VNLAC

| Bacterial strains tested | Mean zone of inhibition ± SEM (in diameter; in mm) | 100 μg/mL | 200 μg/mL | 300 μg/mL | 400 μg/mL | 500 μg/mL | Ciprofloxacin (30 μg disc) | Control (5% DMSO in water) |
|-------------------------|---------------------------------------------------|-----------|-----------|-----------|-----------|-----------|----------------------------|----------------------------|
| S. pyogenes             |                                                   | 11.08 ± 0.24 | 14.05 ± 0.28 | 16.00 ± 0.27 | 17.00 ± 0.24 | 18.00 ± 0.21 | 22.04 ± 0.15 | No zone                     |
| S. aureus               |                                                   | 09.13 ± 0.27 | 12.10 ± 0.28 | 14.07 ± 0.27 | 15.02 ± 0.27 | 16.00 ± 0.24 | 22.00 ± 0.18 | No zone                     |
| E. coli                 |                                                   | 14.07 ± 0.29 | 17.06 ± 0.25 | 20.10 ± 0.28 | 23.11 ± 0.27 | 25.00 ± 0.24 | 28.11 ± 0.12 | No zone                     |
| P. aeruginosa           |                                                   | 13.13 ± 0.29 | 17.02 ± 0.26 | 21.00 ± 0.28 | 22.12 ± 0.24 | 23.04 ± 0.21 | 26.14 ± 0.12 | No zone                     |

SEM is standard error of mean

*Values are reported as mean ± SEM (n = 3); positive control used is ciprofloxacin (standard 30 μg disc)
However, means and methods for improving the efficiency of the VNLAC powder (e.g., by functionalizing or surface modification of VNLAC with suitable chemicals) for the complete inhibition of bacterial growth have to be considered immediately. Nevertheless, the present study bestows with an effective way to produce value-added utilization of phytomass discards in the area of pathogenic science and biomedicine and also paves a way towards green environment. Encouraging results realized from the study would certainly attract researchers globally to study various phytomass wastes-derived carbons as novel therapeutic agents which can effectively hinder the growth of various strains of microbes and further for bio imaging and drug delivery as well. Thus, the work represents an essential attempt in producing low-cost carbon powder from V. negundo plant phytomass that may possess enormous antimicrobial activity also simultaneously dealing recycling/reuse of environmental wastes. Identification of such agents at low cost as well as the feasibility of producing in bulk would further help reduce the economic burden.

**Conclusion**

H₃PO₄-activated carbon powder was produced from the green phytomass waste, namely the leaves of V. negundo L. plant (VNLAC) and the antibacterial efficacy against four human pathogens was evaluated with a view to adopt as an antibacterial in biomedical field. The source of the carbon is thus natural, zero-cost, and renewable one and the synthesis process is also obviously ecobenign. VNLAC contains hetero elements like N, S, and O in the form of various functional groups such as amines, carboxyls, sulphoxide, and alcoholic groups on its surface. These organic functional groups are likely to interact with the cell membrane of the pathogens thereby inhibiting their growth and hence play a vital role in the biological activity of VNLAC. This leaves a wider scope for further research on proposing a plausible mechanism for the antibacterial properties of the...
carbon materials relating the surface organic functional groups.

The present investigation reveals that VNLAC inhibits Gram-negative pathogens efficiently than the Gram-positive ones. The difference in inhibition may be due to the complex cell membrane structure. Further, the antibacterial activity increases as we move from low (100 μg/mL) to high (500 μg/mL) concentration of VNLAC. Obviously, VNLAC may be a prospective antibacterial agent for many Gram-negative and Gram-positive bacteria as well. The carbon precursor as well as the process adopted in achieving the carbon are eco-benign, inexpensive, and can be scaled up. The work presented here can be further extended in developing a series of antimicrobial agents of natural origin and this work explores can be further extended in developing a series of antimicrobial agents of natural origin and this work explores how these are formulated into nanostructured substrates.

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Plant authentication

The biomass was collected from a farm land in Batlagundu, Dindigul District, Tamil Nadu, India. A sample of the leaf bearing voucher Specimen No. NA/MAHU/Chem/001(2018-19) was deposited and preserved at the Department of Botany of one of the reputed Institutes, Tamil Nadu, India, and was identified and authenticated by a botanist/taxonomist of the same institute.

Authors’ contributions

NA collected the phytomass samples and followed the experimental procedures and characterized the samples. PK designed experiments, involved in interpretation of the results, and drafted the manuscript. All authors read and approved the final manuscript.

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