New protocols to access imidazoles and their ring fused analogues: synthesis from N-propargylamines

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Imidazole and its derivatives are privileged N-heterocyclic structures present in various natural products and synthetic pharmaceuticals. Despite the numerous methods that have been developed for the synthesis of imidazole cores, it is still challenging to readily achieve high efficiency and regioselectivity in imidazole synthesis. Therefore, synthesis of these compounds using new protocols is always interesting. In this study we discuss the most representative and interesting reports on the synthesis of imidazoles and their fused analogues from N-propargylamines. Mechanistic aspects of the reactions are considered and discussed in detail.

1. Introduction

Among all of the nitrogen-containing heterocycles, imidazole is one of the most important motifs in organic and medicinal chemistry. Several drug candidates endowed with potent biological activity against cancers, arrhythmias, convulsions, migraines, bacteria, microbes, and viral infections contain an imidazole moiety (Fig. 1). It is also one of the essential cores in a number of ionic liquids and natural products like histidine, histamine, vitamin B12, pilocarpine alkaloids, nucleic acid bases and biotin (Fig. 2). Moreover, imidazole derivatives are widely applied as fungicides for plants and their products.

The preparation of the imidazole cores generally depends on classical approaches, such as the Debus, Radiszewski, Wal-lach, Markwald, and Van Leusen methods. Despite their popularity in academic and industrial areas, these methods suffer from several drawbacks, including low regioselectivity, low yields, and poor functional group tolerance. In this respect,
the design of improved, highly efficient and regioselective approaches for imidazole preparation is of prime importance.

*N*-Propargylamines are some of the most useful and versatile building blocks in organic synthesis because they have diverse reaction patterns. They have been abundantly used as precursors in the synthesis of various nitrogen-based heterocycles and complex natural products. More recently, we published review papers that covered the synthesis of pyrrole, pyridine, quinoline, pyrazine, 1,4-oxazepane, and 1,4-diazepane derivatives from *N*-propargylamines. Over the last ten years, the synthesis of imidazole derivatives from *N*-propargylamines has been a very active area of research. This new protocol in the imidazole synthetic methods offers several advantages, such as high atom economy, high regioselectivity, and shorter synthetic routes. To the best of our knowledge, a comprehensive review on the synthesis of imidazole derivatives from *N*-propargylamines has not been reported in the literature. This study is an attempt to summarize the data available from the literature about the synthesis of imidazoles from *N*-propargylamines (Fig. 3). In this study, we have classified these reactions into two sections. The first section of this review will be focused on the synthesis of substituted imidazoles from *N*-propargylamines. In the second section, methodologies for the synthesis of fused imidazoles from *N*-propargylamines will be discussed. The mechanistic aspects of the reactions will be considered and discussed in detail.

2. Highly substituted imidazoles

One of the earliest reports on the synthesis of the imidazole ring from *N*-propargylamines appeared in 1974 when *N*-propargylamine and acetamidic esters were cyclized to their corresponding imidazoles through the intramolecular cyclization of propargylamidine intermediates. Since then, the use of *N*-propargylamines as attractive starting materials for the synthesis of this heterocyclic system has been widely reported. In 2007, the group of Abbiati showed that a series of 4-substituted-2-phenylimidazoles were formed from easily available *N*-propargyl-benzamidine and aryl halides through a tandem aminopalladation/reductive elimination/isomerization method employing Pd(PPh3)4 as the catalyst, CuI as the co-catalyst, and K2CO3 as the base, in anhydrous DMF. They found that the presence of a co-catalyst was crucial for the success of this reaction. In the absence of CuI, low reaction yields and long reaction times were observed. Depending on the electronic effects of substituents on the aryl ring, substrates with electron-withdrawing groups gave higher yields than those with electron-donating groups. The reaction also worked well with heteroaryl halides (Scheme 1).

Along this line, Wu and co-workers reported a Pd(II)-catalyzed coupling reaction of fluorinated propargylamines with aryl iodides into 2-fluoroalkyl-5-benzyl imidazoles (Scheme 2). After studying a number of solvents, catalysts, and bases,
a combination of Pd(OAc)2/K2CO3/DMF at 80 °C was found to be optimal with respect to isolated product yield. The optimized protocol tolerated a variety of functional groups, such as chloro, fluoro, methoxy, acetyl, and ester functionalities. However, the substrate with R1 = CF2Br failed to participate in this reaction. In contrast to Abbiati’s method, this method was more efficient for aryl halides bearing an electron-donating group.

An important study on imidazole-5-carbaldehydes was carried out by the same research team, in which propargylamidines were transformed into substituted imidazole-5-carbaldehydes upon treatment with N-iodosuccinimide (NIS) in the presence of Ph3PAuCl/AgBF4/K2CO3 as a catalytic system in acetone at room temperature (Scheme 3a). The nature of the substituent attached to the aryl ring had a small impact on the success of the cyclization. Under optimized conditions, the reaction tolerated both electron-donating and electron-withdrawing substituents on the aryl ring and gave final products in good to excellent yields. However, internal alkynes failed to participate in the reaction. The mechanistic course of this reaction sequence is shown in Scheme 3b, and involves the initial formation of intermediate A from the reaction of activated propargylamidine 7 and NIS. The hydrogen shift of intermediate A gives intermediate B, which undergoes homolytic cleavage at the C–I bond to produce radical intermediate C. Subsequently, reaction of this intermediate with air oxygen gives peroxy-intermediate D. Finally, removal of a hydroxyl radical from D affords the observed imidazole-5-carbaldehydes.

In 2010, Shen and Xie described a general and efficient synthesis of a diverse collection of di- and tri-substituted imidazoles via the Ti-catalyzed regioselective cyclization of the corresponding N-propargylamines with nitriles under reflux conditions in toluene. The results demonstrated that the substrates with an internal alkyne unit gave higher yields than those with a terminal unit, and aryl nitriles were more reactive than alkyl nitriles. It was also found that the electronic character of the aryl nitriles had little to no effect on the yield or outcome of the methodology. The mechanism proposed by the authors to explain this reaction starts with the generation of the titanacarborane amide via the amine-exchange reaction between [(Me3NCH2)2C2B9H9]Ti(NMe2)9 and N-propargylamine 10. Then, coordination of nitrile 11 to the Ti atom of this intermediate furnishes intermediate A.

Fig. 1 Selected examples of drugs containing an imidazole core.

Fig. 2 Natural sources of imidazoles.
Subsequently, a migratory insertion gives intermediate C. The formation of intermediate D occurs next, followed by its acid–base reaction with N-propargylamine 10 to give the dihydroimidazole E with concomitant regeneration of A. Finally, the dehydrative aromatization of E affords the observed products 12 (Fig. 4). Recently the authors improved the efficiency of this reaction in terms of the yield and reaction time by performing the process in THF using Sm[N(SiMe$_3$)$_2$]$_3$ as the catalyst.

The same authors were also able to demonstrate that a series of 2-aminoimidazoles 15 could be obtained from the

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**Scheme 1** Synthesis of 4-substituted-2-phenylimidazoles 3 via a Pd-catalyzed cascade reaction of N-propargyl-benzamidine 1 and aryl halides 2.

**Scheme 2** Pd(II)-Mediated synthesis of 2-fluoroalkyl-5-benzyl imidazoles 6 developed by Wu.
reaction of N-propargylamines 13 with carbodiimides 14 in the presence of 5 mol% titanacarborane monoamide 9 in refluxing toluene (Scheme 4a). The reaction tolerated both terminal and aryl-substituted internal N-propargylamines and gave the corresponding 2-aminoimidazoles in good to excellent yields; however, extension of the reaction to alkyl-substituted

**Scheme 3** (a) Au(I)-Catalyzed intramolecular hydroamination of the fluorinated N-propargylamidines 7 to imidazole-5-carbaldehydes 8. (b) Proposed mechanism for the formation of 8.

**Fig. 4** Mechanism proposed to explain the synthesis of substituted imidazoles 12 developed by Shen and Xie.
internal alkynes failed. The reaction proceeds along a similar mechanistic pathway to that described in Fig. 4. It is worth noting that when secondary propargylamines were used as starting materials, instead of 2-aminoimidazole 15, the corresponding N-(1H-imidazol-2(3H)-ylidene)amines 16 were formed as the final products because they could not undergo aromatization to give imidazoles 15 (Scheme 4b).27

More recently, Wang and co-workers studied the possibility of synthesizing imidazoles 19 from N-propargylamines 17 and ketenimines 18 through a silver-catalyzed nucleophilic addition/5-exo-dig cyclization/isomerization sequential process. The optimal reaction conditions developed included stirring of 1 mmol of the propargylamine, 1 equiv. of the ketenimine, 1 equiv. of Et3N and 10 mol% of AgOTf in 10 mL of THF at reflux temperature. The optimized protocol tolerated a variety of functional groups, such as nitro, bromo, ester, and methoxy, and generally provided the corresponding 1,2,5-trisubstituted 1H-imidazoles 19 in good yields (Scheme 5). The author proposed mechanism for this cyclization, which is outlined in Scheme 6.9

An efficient route to afford tetrasubstituted imidazoles 23 has been developed by the same group, which was based on a one-pot, two-step, three-component reaction between N-propargylamines 20, terminal alkynes 21, and N-sulfonyl azides 22 (Scheme 7a). These steps involve: (i) stirring of the starting materials in the presence of CuI as a catalyst and Et3N as a base in acetonitrile under nitrogen atmosphere and (ii) addition of Cs2CO3 and heating of the reaction mixture at 80 °C for 4 h. According to the mechanistic studies, this reaction proceeds through the Cu-catalyzed formation of ketenimine intermediate A, from the starting alkynes 21 and N-sulfonyl azides 22, followed by the nucleophilic addition of N-propargylamine 20 to the central carbon of ketenimine A to give intermediate B, which, in the presence of base,
transforms into allene intermediate C. Subsequently the 6π-electrocyclic ring closing of intermediate C gives the ionic intermediate D. After a sulfonyl 1,3-shift, imidazoles 23 are observed (Scheme 7b).28

3. Fused imidazoles

3.1. Imidazo[1,2-a]pyridines

Despite the fact that intramolecular cyclization of 2-amino N-propargylpyridinium bromides to afford the corresponding imidazo[1,2-a]pyrimidines had been known since 1964,29 it was not until 2008 that Bakherad and co-workers developed the first intramolecular cyclization of 2-amino N-propargylpyridinium bromides to form imidazo[1,2-a]pyridines. They showed that treatment of N-propargylpyridinium bromide 24 with iodobenzenes 25 in the presence of [(PPh3)2PdCl2]Cu/Et3N as a catalytic system in DMF resulted in the corresponding 2-benzylimidazo[1,2-a]pyridines 26 in good to high yields. However, iodobenzenes bearing an electron-donating group failed to participate in this reaction. This transformation is believed to occur through a tandem Sonogashira coupling/intramolecular cyclization/aromatization sequential process (Scheme 8a).30 Shortly afterwards, the authors improved their methodology in terms of yield and reaction time using polystyrene-supported palladium(II) ethylenediamine complex 27 as the catalyst (Scheme 8b).31 This reaction has been successfully applied as the key strategic step in synthesis of a series of
imidazopyridinium analogues 28 as antagonists of neuropeptide S receptor (Scheme 9). More recently, an impressive and more robust protocol for the synthesis of substituted imidazo[1,2-a]pyridines was introduced by Nguyen et al. They found that readily accessible N-proparglypyridiniums 29, in the presence of NaOH in water, rapidly cyclized and generally produced the corresponding substituted 2-methylH-imidazo[1,2-a]pyridines 30 in almost quantitative yields (Scheme 10). The mechanism of this cyclization was demonstrated via isotopic labeling experiments. The mechanism has been determined to proceed via a tandem alkyne-allene isomerization/Nazarov-type cyclization.

The possibility of regioselective intramolecular cyclization of N-propargylaminopyridines to pyridine fused imidazoles was first realized by Savic and co-workers, who synthesized a series of mono and disubstituted 3-methylH-imidazo[1,2-a]pyridine derivatives 32 from Boc-protected N-propargylaminopyridines 31 in the presence of tBuOK in THF. As shown in Scheme 11, the reaction showed remarkable flexibility and the desired products were formed in moderate to good yields with both electron-rich and electron-poor N-propargylaminopyridines; however, it could not be extended to 3-bromo substituted pyridines. Shortly after, the group of M. Chioua further improved the efficiency of this cyclization using AgOTf as the catalyst in deoxygenated acetonitrile. They also probed the mechanism of the reaction and found that the reaction proceeded via a 5-exo-dig cyclization process.

With the objective of designing a greener procedure to pyridine-fused imidazoles through an intramolecular hydroamination strategy, Adimurthy and co-workers were able to demonstrate that a series of substituted imidazo[1,2-a]pyridines 34 could be obtained from N-propargylaminopyridines 33 under metal-free conditions in the most environmentally benign solvent, water. The mechanistic course of this reaction sequence is depicted in Scheme 12 and shows that water, presumably, plays a dual role as solvent and catalyst. It should be mentioned that the reaction failed under an open air atmosphere.

Interestingly, when the abovementioned reaction is performed in the presence of AgNO3/TMEDA/acetone in an oxygen atmosphere, imidazo[1,2-a]-pyridine-3-carbaldehydes 36 are obtained. The mechanism proposed to explain this
transformation starts with the generation of π-complex A via coordination of the alkyne moiety with silver. Then, the cyclization of A gives intermediate B. Subsequently, addition of oxygen to intermediate B furnishes organosilver peroxide intermediate C. The formation of intermediate D occurs next, followed by its isomerization to give intermediate E. Finally, the elimination of the silver(II) species affords the observed products (Scheme 13).36 It should be noted that, previously, the research team of Marco-Contelles had reported two examples of imidazo[1,2-α]-pyridine-3-carbaldehydes preparation via the Sandmeyer reaction of N-propargylanopropiridines.37

3.2. Imidazo[1,2-α]-pyrazine (piperazine)
An interesting and rare example for the synthesis of piperazine-fused imidazole was reported by McCort and Pascal in 1992. In anhydrous toluene at 100°C, an intermolecular heterocyclization between 3-ethoxy-tetrahydropyrazine 37 and N-propargylamine 38 furnished imidazo[1,2-α]piperazine 39 in a yield of 65% (Scheme 14a).39 Later, imidazo[1,2-α]quinoxaline 41 was synthesized via an intramolecular reaction (Scheme 14b). 4-Amino-substituted analogues of 41 showed good inhibitory activities against IKK (IκB kinase).40

A new method for the synthesis of substituted benzimidazolopyrazines 44 from N-propargylamine 42 and phenylenediamines 43, based on a tandem imidazole formation/heteroannulation process, has been reported recently. The reaction takes place through a tandem process consisting of an initial imidazole formation, followed by a copper-catalyzed 6-exo-dig cyclization and final isomerization (Scheme 15).41

3.3. Imidazo-diazepines
One of the earliest reports on the synthesis of benzoazepine-fused imidazoles from N-propargylamines appeared in 1977 when
4-amino-benzo[b][1,4]diazepinone 45 underwent cyclization with N-propargylamine 46 in the presence of p-toluene sulfonic acid as a catalyst in boiling butanol. The corresponding benzo[b][1,4]diazepinone-fused imidazoles 47 were obtained in moderate to high yields (Scheme 16). It is noted that 4-ethylthio-benzo[b][1,4]diazepines also gave the corresponding diazepine-fused imidazoles using the aforementioned reaction conditions.

In 2005, the group of Gracias described a general and efficient synthesis of tetracyclic diazepine-fused imidazoles 52 from N-propargylamines 48, o-azidobenzaldehydes 49, and p-toluene-sulfonylmethyl isocyanides 50 through a van Leusen/alkyne–azide cycloaddition sequence. The reaction starts with the formation of van Leusen intermediates 51 using K$_2$CO$_3$ as a catalyst in DMF, and then, in situ intramolecular alkyne–azide cycloaddition of intermediates 50 furnishes the corresponding diazepine-fused imidazoles 52 in good to high yields. In the case of internal alkynes, the intramolecular alkyne–azide cycloaddition step was performed in tBuOH/H$_2$O (1:1), and CuSO$_4$·5H$_2$O/sodium ascorbate was used as the catalytic system (Scheme 17).

A straightforward route to 9H-benzo[f]imidazo[1,2-d:1,2,3]triazolo[1,5-a][1,4]diazepines 56, which were prepared because of their potential pharmaceutical interest, was developed by Kurth and co-workers. This route was based on a one-pot four-component reaction in methanol using o-azidobenzaldehydes 53, α-diketones 54, N-propargylamines 55, and ammonium acetate in the presence of 10 mol% InCl$_3$ as the catalyst (Scheme 18). The mechanism of this cyclization reaction is believed to involve: (i) coordination of InCl$_3$ to the oxygen of the aldehyde, which resulted in 2017. RSC Advances Open Access Article. This journal is © The Royal Society of Chemistry 2017 RSC Advances Review Open Access Article. Published on 20 January 2017. Downloaded on 1/20/2025 3:47:33 AM. This article is licensed under a Creative Commons Attribution 3.0 Unported Licence. View Article Online.
in the formation of intermediate $A$; (ii) nucleophilic addition of ammonia to activated aldehyde $A$ to produce imine intermediate $B$; (iii) reaction of imine $B$ with $N$-propargylamine to generate (2-azidophenyl)-$N$-(prop-2-ynyl)methanediamine intermediate $C$; (iv) intermolecular cyclization of $C$ with activated $\alpha$-diketone $D$ to give intermediate $E$; and (v) intramolecular [3 + 2] Huisgen cycloaddition of $E$ produced the corresponding imidazo-diazepines $56$. In another possible mechanism, intramolecular [3 + 2] Huisgen cycloaddition of intermediate $C$ affords intermediate $F$ that transforms to the observed product $56$ via condensation with activated $\alpha$-diketone $D$ (Scheme 19).

### 3.4. Purine-fused imidazoles

The intramolecular cyclization of $N$-propargyladenines $57$, in the presence of CuBr/CsCO$_3$/n-Bu$_4$NBr as a catalytic system, to give purine-fused imidazoles $58$ was described by Qu and co-workers in 2014. They observed that the substituent effects on the intramolecular cyclization afforded a mixture of products. Thus, the terminal alkynes afforded predominantly purine-fused imidazoles $58$, but the aryl substituted internal alkynes yielded a mixture of $58$ and dihydroimidazoles $59$. The substrates that had electron-rich aryl rings in the alkenyl part favored endocyclic double bond products, whereas the substrates bearing electron-poor aryl rings in the alkyne terminus favored exocyclic double bond products (Table 1). According to the proposed mechanism, the reaction proceeds via an alkyne–allene isomerization/tautomerization/cyclization sequential process.

### Table 1  Cu(I)-Catalyzed intramolecular cyclization of $N$-propargyladenines $57$

| Entry | Ar         | Yield (%) | $58 : 59$ |
|-------|------------|-----------|-----------|
| 1     | Ph         | 88        | 99 : 1    |
| 2     | 2-Me-Ph    | 89        | 92 : 8    |
| 3     | 3-Me-Ph    | 87        | 91 : 9    |
| 4     | 3,5-Di-Me-Ph| 79        | 83 : 13   |
| 5     | 2-Ome-Ph   | 85        | 86 : 14   |
| 6     | 4-Ome-Ph   | 85        | 88 : 12   |
| 7     | 4-OEt-Ph   | 88        | 92 : 8    |
| 8     | 3-Cl-Ph    | 81        | 47 : 53   |
| 9     | 4-Cl-Ph    | 88        | 45 : 55   |
| 10    | 4-F-Ph     | 75        | 43 : 57   |
| 11    | 2-Thiophyl | 77        | 45 : 55   |

### 4. Conclusion

The imidazole scaffold is a core structure of a large number of natural products and synthetic pharmaceuticals that show an
impressive variety of biological properties. Such compounds exhibit anticancer, anti-tubercular, anti-parasitic, anti-neuropathic, anti-convulsion, anti-Parkinson’s, antifungal, antibacterial, antiviral, antihistaminic, anti-inflammatory, and anti-obesity properties. Several commercially available drugs, including alpiderm, zolpidem, clonidine, metronidazole, flumazenil, bretazenil, imidazenil, and midazolam, are derived from imidazo-core entities. Consequently, numerous methods have been devoted to the design of expedient and efficient synthetic routes to this heterocycle. Some of the most popular methods for their preparation include the Debos, Radiszewski, Wallach, Marckwald, and Van Leusen reactions. Widespread synthetic routes to this heterocycle. Some of the most popular methods for the generation of fused imidazole cores is particularly interesting.

New methods that produce complex molecules from simpler materials in a single operation are important challenges for modern synthetic chemistry. Over the past decade, N-propargylamines have attracted much attention due to the fact that they are versatile and simple synthetic intermediates for the synthesis of imidazole derivatives from imidazo-core entities. Consequently, numerous e-scientific methods for the use of these methods is limited by the low yields, low selectivity, or narrow substrate scope. Thus, new and more efficient synthetic methods are sought. Developing more efficient methods for the generation of fused imidazole cores is particularly interesting.

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