New Series of Potent Allosteric Inhibitors of Deoxyhypusine Synthase

Yuta Tanaka,* Osamu Kurasawa, Akihiro Yokota, Michael G. Klein, Bunai Saito, Shigemitsu Matsumoto, Masanori Okaniwa, Geza Ambrus-Aikelin, Noriko Uchiyama, Daisuke Morishita, Hiromichi Kimura, and Shinichi Imamura

ABSTRACT: Deoxyhypusine synthase (DHPS) is the primary enzyme responsible for the hypusine modification and, thereby, activation of the eukaryotic translation initiation factor 5A (eIF5A), which is key in regulating the protein translation processes associated with tumor proliferation. Although DHPS inhibitors could be a promising therapeutic option for treating cancer, only a few studies reported druglike compounds with this inhibition property. Thus, in this work, we designed and synthesized a new chemical series possessing fused ring scaffolds designed from high-throughput screening hit compounds, discovering a 5,6-dihydrothieno[2,3-c]pyridine derivative (26d) with potent inhibitory activity; furthermore, the X-ray crystallographic analysis of the DHPS complex with 26d demonstrated a distinct allosteric binding mode compared to a previously reported inhibitor. These findings could be significantly useful in the functional analysis of conformational changes in DHPS as well as the structure-based design of allosteric inhibitors.

KEYWORDS: Deoxyhypusine synthase (DHPS), eukaryotic translation initiation factor 5A (eIF5A), nicotinamide adenine dinucleotide (NAD), spermidine, allosteric inhibitor

Deoxyhypusine synthase (DHPS) is an enzyme that activates via hypusine \([\text{N}^\text{ε}(4\text{-amino-2-hydroxybutyl})-\text{lysine}]\) modification the eukaryotic translation initiation factor 5A (eIF5A), which is a regulator of the protein translation processes involved in tumor growth.\(^1\)\(^-\)\(^5\) This modification involves two enzymatic steps. DHPS catalyzes the first rate-limiting step, which is the cleavage of the polyamine spermidine: the 4-aminobutyl moiety is transferred to the \(\varepsilon\) amino group of one specific lysine residue of the eIF5A precursor to form a deoxyhypusine intermediate.\(^6\)\(^-\)\(^8\) In the second step, deoxyhypusine hydroxylase (DOHH) converts the deoxyhypusine-containing intermediate into hypusine-containing mature eIF5A (Supporting Information). Blocking the biological activity of eIF5A by inhibiting the DHPS-mediated modification of its precursor could be an effective strategy in cancer therapy.\(^9\)\(^,\)\(^10\) \(N^1\)-Guanyl-1,7-diaminoheptane (GC-7) (1), a substrate (spermidine) mimic, binds to DHPS similarly to spermidine\(^11\)\(^-\)\(^13\) and possesses inhibitory activity; nonetheless, the potential application of 1 is limited by its polyamine-like structure (Figure 1A).\(^14\)\(^-\)\(^16\) We recently reported a novel allosteric inhibitor (2) and its cocrystal structure with DHPS, which revealed the dynamic conformational change in the DHPS structure.\(^17\) However, its application for the further biological evaluation of DHPS is still limited due to the narrow structure–activity relationship (SAR) of its related compounds, although the structural analysis suggests that the allosteric pocket is druggable. Hence,
more potent and druglike inhibitors are needed for detailed biologic studies on the therapeutic potential of targeting the DHPS-mediated hypusine synthesis. Here, we describe a new series of potent DHPS inhibitors discovered through SAR studies from hit compounds obtained via a high-throughput screening (HTS). Among them, the X-ray crystallographic analysis of a highly potent compound (26d, IC\textsubscript{50} = 0.0092 \mu M) in complex with DHPS demonstrates a distinct binding mode compared to the previously reported 2.

To identify a new series of DHPS inhibitors, we conducted an HTS via an enzyme assay. Several hit compounds containing a 3-pyridyl group such as 3 and 4 (Figure 1B), whose chemical structures differ from those of other reported inhibitors (Figure 1A), showed inhibitory activities toward DHPS.\textsuperscript{17−20} An initial SAR study revealed that the 3-pyridine moiety is essential for their inhibitory activity (data not shown). Based on the structural similarity among these compounds, we presumed that a fused ring system with 3-pyridine is a pharmacophore of their series for inhibiting DHPS, as shown in Figure 1B; thus, we started chemical modifications according to the designed molecules maintaining the possible pharmacophore.

Benzyl derivatives with designed fused rings 6a–d were synthesized via benzylation of 5a–d as illustrated in Scheme 1. The precursors 5a–d were prepared as described in the Supporting Information.

Scheme 1. Synthesis of Benzyl Derivatives 6a–d\textsuperscript{a}

"Reagents and conditions: (a) benzyl bromide, NaH, DMF.

Scheme 2 illustrates the synthesis of 5,6-dihydrothieno[2,3-\text{c}]pyridine derivatives 9a–g from the intermediate 5d. Alkylation of 5d with methyl 2-bromo-2-phenylacetate gave the racemic ester 7a, which was successively reduced by NaBH\textsubscript{4} in the presence of CaCl\textsubscript{2} to yield the alcohol 8a. Conversion into the racemic amine 9a was performed via introduction of phthalimide by Mitsunobu reaction and subsequent treatment with hydrazine monohydrate at 70 °C in EtOH. The optical resolution of 9a via chiral HPLC

"Reagents and conditions: (a) methyl 2-bromo-2-phenylacetate, potassium tert-butoxide, THF, 90%; (b) NaBH\textsubscript{4}, CaCl\textsubscript{2}, MeOH; (c) (i) phthalimide, triphenylphosphine, diisopropyl azodicarboxylate, toluene; (ii) hydrazine monohydrate, EtOH, 70 °C; (d) 10a–b, potassium tert-butoxide, THF; (e) K\textsubscript{2}CO\textsubscript{3}, MeOH, H\textsubscript{2}O, 80 °C, 62%; (f) 11, potassium tert-butoxide, DMF, 52%; (g) 2 M NaOH, MeOH, 87%.
afforded both its enantiomers: the S- (9b) and the R-form (9c). The absolute configuration of each enantiomer was determined through single-crystal X-ray analysis (Supporting Information). Using the same procedures for 9a, compounds 9d−e were synthesized from 5d and the corresponding bromides 10a−b, which were prepared as described in the Supporting Information. The Boc group of 9e was removed in the presence of K2CO3 at 80 °C in MeOH and H2O to obtain the indole derivative 9f. By treating 5d with the bromide 11, we formed the benzoyl derivative 12, which was hydrolyzed to yield the alcohol 13, which was then converted into the amine 9g similarly to 9a.

Scheme 3 describes an alternative method to obtain 5,6-dihydrothieno[2,3-c]pyridine derivatives. By using 3-hydroxythiophene 14 as a starting material, trifluoromethanesulfonic anhydride, pyridine; (d) (i) (E)-1-ethoxyethene-2-boronic acid pinacol ester, Pd(dppf)Cl2·CH2Cl2, Cs2CO3, DME, H2O, 80 °C, 72%; (c) trifluoromethanesulfonic anhydride, pyridine; (d) (i) (E)-1-ethoxyethene-2-boronic acid pinacol ester, Pd(dppf)Cl2·CH2Cl2, Cs2CO3, DME, H2O, 80 °C; (ii) 12 M HCl, THF, 30% (two steps); (e) 19a−b, 2-picoline borane, MeOH, AcOH; (f) NaOMe, MeOH; (g) 4 M HCl in EtOAc; (h) 23a−d, 2-picoline borane, MeOH, AcOH; (i) (i) phthalimide, triphenylphosphine, disopropyl azodicarboxylate, toluene; (ii) hydrazine monohydrate, EtOH, 70 °C.

Reagents and conditions: (a) bromine, AcOH, 61%; (b) 3-pyridineboronic acid, Pd(dppf)Cl2·CH2Cl2, Cs2CO3, DME, H2O, 80 °C, 72%; (c) trifluoromethanesulfonic anhydride, pyridine; (d) (i) (E)-1-ethoxyethene-2-boronic acid pinacol ester, Pd(dppf)Cl2·CH2Cl2, Cs2CO3, DME, H2O, 80 °C; (ii) 12 M HCl, THF, 30% (two steps); (e) 19a−b, 2-picoline borane, MeOH, AcOH; (f) NaOMe, MeOH; (g) 4 M HCl in EtOAc; (h) 23a−d, 2-picoline borane, MeOH, AcOH; (i) (i) phthalimide, triphenylphosphine, disopropyl azodicarboxylate, toluene; (ii) hydrazine monohydrate, EtOH, 70 °C.

To verify the potentials of each bicyclic scaffold having the 3-pyridyl moiety, the derivatives with a benzyl group derived from the hit compound 4 were evaluated in the enzyme assay (Table 1). The conversion of the thieno[3,2-d]pyrimidine ring in 4 to isoquinolone 6a improved the inhibitory activity (IC50 = 8.7 μM). Dihydroisoquinolone 6b also showed a comparable potency to 6a, indicating that the left ring of the bicyclic scaffold is not necessarily aromatic and the nitrogen atom on the 1-position of the thieno[3,2-d]pyrimidine (4) weakens the potency. A significant loss of potency was observed by changing the position of the 3-pyridine moiety (6c, IC50 > 100 μM), indicating that the orientation of the substituents from the scaffold is a key factor. Furthermore, the evaluation of scaffolds such as fused the 6,5- (6d) and 5,6-membered (27) bicyclic rings revealed the strong inhibitory activity of 5,6-dihydrothieno[2,3-c]pyridine 6d (IC50 = 6.3 μM).
Based on the above-mentioned SAR results, we further explored the substituents on the left side of the 5,6-dihydrothieno[2,3-c]pyridine scaffold (Table 2). Since the hit compound 3 bearing a hydroxymethyl group at the benzyl position showed inhibitory activity, we expected that the introduction of a polar functional group at the corresponding benzylic position of 6d would be tolerable. Indeed, introduction of the hydroxymethyl group (8a) significantly enhanced the potency although a methyl ester (7a) was unfavorable, indicating the necessity of a hydrogen donor for the inhibitory activity. The conversion of the primary alcohol to an amine further improved the potency, and the racemate amine 9a exhibited a strong IC50 value of 0.018 μM. Moreover, we found that R-form 9c is a eutomer after separation of each enantiomer of 9a. On the other hand, neither the extension of the alkyl chain (9g) nor the conformational restriction of the amine moiety via cyclization (22a) increased the potency compared to 9a.

In a previous work, the allosteric inhibitor 2 showed nicotinamide adenine dinucleotide (NAD), a coenzyme of DHPS, competitive profile as a result of dramatic conformational changes of DHPS induced by its binding.17 We employed the radiometric assay using [3H]-spermidine to evaluate the inhibitory potency of compounds and the competitiveness against NAD or spermidine at each Km or higher concentrations although there is a limitation to analyze precise kinetic profiles due to 120 min incubation time needed in this assay. To analyze the inhibition mode of the newly discovered chemical series, the potent compound 9c was evaluated under the high NAD concentration (Table 3); its DHPS inhibition at 250 μM NAD was 29-fold lower than at 14 μM NAD, indicating a NAD competitive profile similar to that of 2.17 The X-ray cocrystal analysis of the new chemical series revealed its binding mode and a mechanism of the NAD competition (vide infra).

To further improve the inhibition potency and identify a promising tool molecule, we successively converted the benzene ring on the left part into other hydrophobic substituents (Table 3). We used a DHPS concentration of

Table 1. SAR for Fused Ring Scaffolds 6a−d and 27

| compd | Scaffold | DHPS IC50 (μM) |
|-------|---------|---------------|
| 4     | (HTS hit) | 66            |
| 6a    |         | 8.7           |
| 6b    |         | 8.2           |
| 6c    |         | >100          |
| 6d    |         | 6.3           |
| 27    |         | 53            |

*IC50 values were calculated via the nonlinear regression analysis of the percent inhibition data. 95% confidence intervals are shown in the Supporting Information.

Table 2. SAR for Substituents of 5,6-Dihydrothieno[2,3-c]pyridine-7(4H)-one Derivatives

| compd | R† | DHPS IC50 (μM) |
|-------|----|---------------|
| 6d    | Bn | 6.3           |
| 7a    |    | >100          |
| 8a    |    | 0.15          |
| 9a    |    | 0.018         |
| 9b    |    | 0.16          |
| 9c    |    | 0.014         |
| 9g    |    | 0.44          |
| 22a   |    | 16            |

*Refer to Table 1. †Racemate.
Since the tight-binding possibly caused an underestimation of the IC50 value of a highly potent compound, we analyzed the inhibitory activities also under high NAD concentrations to obtain weakened IC50 values for the SAR analysis of the NAD competitive inhibitors. The conversion of the benzene ring (9c) into bicyclic rings such as benzothiophene (9d) and indole (9f) decreased the inhibitory activity, while the compound with a cyclohexane substituent (26a) exhibited improved potency under high NAD concentration (IC50 = 0.17 μM). This result encouraged us to pursue other alkyl substituents. Although n-hexane derivative 22b also showed potent activity, its metabolic stability in mice liver microsome was diminished, probably due to high lipophilicity. Therefore, we examined shorter and/or branched alkyl chains (26b–d) to reduce the lipophilicity, resulting in the discovery of the isobutyl derivative 26d, which exhibited the most potent inhibitory activity among the 5,6-dihydrothieno[2,3-c]pyridine derivatives (IC50= 0.069 μM under high NAD condition).

Table 3. Conversion of Benzene Ring to Other Hydrophobic Substituents

| compd | R² | DHPS IC50 (μM)a | MLMb (μL/mi, n/mg) | clogPc |
|-------|----|----------------|-------------------|--------|
| 2     | -  | 0.062 / 56     | 304               | 3.21   |
| 9c    |    | 0.014 / 0.41   | 71                | 2.54   |
| 9d    |    | 0.029 / 0.63   | 185               | 3.57   |
| 9f    |    | 0.11 / N.D.    | 59                | 2.53   |
| 26a   |    | 0.012 / 0.17   | 59                | 3.14   |
| 22b   |    | 0.030 / 0.58   | 114               | 3.67   |
| 26b   |    | 0.026 / 0.79   | 32                | 2.08   |
| 26c   |    | 0.011 / 0.23   | 37                | 2.35   |
| 26d   |    | 0.0092 / 0.069 | 44                | 2.48   |

aRefer to Table 1. bThe metabolism clearance of each compound was examined by using mice liver microsomes. cDetermined by using Daylight software. dRacemate. eNot determined.

Table 4. Pharmacokinetic Parameters of 26d in Micea

| Vdss (mL/kg) | CLtotal (mL/h/kg) | Cmax (ng/mL) | AUClc0–8h (ng-h/mL) | MRT (h) | F (%) |
|--------------|-------------------|--------------|---------------------|---------|-------|
| 8491         | 8965              | 4.8          | 13.1                | 2.46    | 11.7  |

aCassette dosing. Male Institute of Cancer Research mice (n = 3). b0.1 mg/kg, iv. c1 mg/kg, po. dBioavailability.

0.024 μM in the enzyme assay (Supporting Information). Since the tight-binding possibly caused an underestimation of the IC50 value of a highly potent compound, we analyzed the inhibitory activities also under high NAD concentrations to obtain weakened IC50 values for the SAR analysis of the NAD competitive inhibitors. The conversion of the benzene ring (9c) into bicyclic rings such as benzothiophene (9d) and indole (9f) decreased the inhibitory activity, while the compound with a cyclohexane substituent (26a) exhibited improved potency under high NAD concentration (IC50 = 0.17 μM). This result encouraged us to pursue other alkyl substituents. Although n-hexane derivative 22b also showed potent activity, its metabolic stability in mice liver microsome was diminished, probably due to high lipophilicity. Therefore, we examined shorter and/or branched alkyl chains (26b–d) to reduce the lipophilicity, resulting in the discovery of the isobutyl derivative 26d, which exhibited the most potent inhibitory activity among the 5,6-dihydrothieno[2,3-c]pyridine derivatives (IC50= 0.069 μM under high NAD condition).

In addition, 26d showed orally available pharmacokinetic profiles in mice (Table 4). Further investigations are needed to verify
Another NAD binding site and hinders its pocket. To our surprise, this Lys287 is overlapped with the spermidine binding site in the DHPS/2 complex structure (Figure 2B, C). Overall, these results indicate the versatility of the allosteric site as a druggable pocket, which can be utilized for exploring DHPS inhibitors.

Additionally, this complex structure well explained the results of the SAR studies (Figure 3). Regarding the 3-pyridyl moiety, a hydrogen bonding was observed between a nitrogen atom of the pyridine ring and a side chain of Asp262 and/or a main chain of Ser233 via a water molecule, confirming that it is required for the potency. Furthermore, the carbonyl oxygen on the bicyclic scaffold formed a hydrogen bond with Lys287. The primary amine of 26d also formed a hydrogen bond with Asp243 (side chain) and Asp238 (main chain), resulting in high affinity in the pocket and strong inhibitory activity. The alkyl side chain occupies a hydrophobic pocket surrounded by Phe100, Ala270, Met278, and Val286; however, there might be other residues surrounding the alkyl side chain because the 288–298 and 316–334 residues are disordered in the 26d complex. As described in Table 1, the orientation of substituents on the fused ring scaffolds is important for the potency, which is reasonable to maintain or stabilize the interactions between inhibitor and DHPS.

In summary, we discovered a new class of potent allosteric DHPS inhibitors via SAR studies by initially exploring bicyclic scaffolds suggested by the pharmacophore of HTS hit compounds. These compounds were characterized as NAD competitive inhibitors, and the X-ray cocystal analysis revealed that the potent inhibitor 26d binds to the allosteric pocket in a newly identified binding mode, distinct from that of the previously reported inhibitor 2. The identification of the novel binding mode with the compound possessing improved enzymatic inhibitory activity and favorable PK properties suggests the promise of using the allosteric site as a druggable pocket, and these results would pave the road for the structure-based design for further lead optimization of allosteric inhibitors targeting DHPS.

**ASSOCIATED CONTENT**

**Supporting Information**

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsmedchemlett.0c00331.

**Figure of the DHPS-eIF5A pathway and chemical reaction of DHPS, 95% confidence intervals of enzyme IC_{50}, synthetic experimental details and characterization data, single-crystal X-ray structures of 9b and 9c, assay protocols, and cocystal data of 9c and 26d (PDF)**

**Accession Codes**

The atom coordinates and structure factors for the DHPS/9c and DHPS/26d complexes have been deposited in the Protein Data Bank with the accession codes 6WKZ, and 6WL6, respectively.

**AUTHOR INFORMATION**

**Corresponding Author**

Yuta Tanaka — Pharmaceutical Research Division, Takeda Pharmaceutical Company Limited, Fujisawa, Kanagawa 251-8555, Japan; orcid.org/0000-0002-6110-7990; Phone: +81-466-32-1164; Email: yuta.tanaka@takeda.com

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Authors

Osamu Kurasawa — Pharmaceutical Research Division, Takeda Pharmaceutical Company Limited, Fujisawa, Kanagawa 251-8555, Japan

Akihiro Yokota — Pharmaceutical Research Division, Takeda Pharmaceutical Company Limited, Fujisawa, Kanagawa 251-8555, Japan

Michael G. Klein — Department of Structural Biology, Takeda California, San Diego, California 92121, United States

Bunnai Saito — Pharmaceutical Research Division, Takeda Pharmaceutical Company Limited, Fujisawa, Kanagawa 251-8555, Japan

Shigemitsu Matsumoto — Pharmaceutical Research Division, Takeda Pharmaceutical Company Limited, Fujisawa, Kanagawa 251-8555, Japan

Masanori Okaniwa — Pharmaceutical Research Division, Takeda Pharmaceutical Company Limited, Fujisawa, Kanagawa 251-8555, Japan

Geza Ambrus-Aikelin — Department of Structural Biology, Takeda California, San Diego, California 92121, United States

Noriko Uchiyama — Pharmaceutical Research Division, Takeda Pharmaceutical Company Limited, Fujisawa, Kanagawa 251-8555, Japan

Daisuke Morishita — Pharmaceutical Research Division, Takeda Pharmaceutical Company Limited, Fujisawa, Kanagawa 251-8555, Japan

Hiromichi Kimura — Pharmaceutical Research Division, Takeda Pharmaceutical Company Limited, Fujisawa, Kanagawa 251-8555, Japan

Shinichi Imamura — Pharmaceutical Research Division, Takeda Pharmaceutical Research Division, Takeda Pharmaceutical Research Division, Takeda Pharmaceutical Research Division, Takeda Pharmaceutical Research Division

Complete contact information is available at: https://pubs.acs.org/10.1021/acsmedchemlett.0c00331

Author Contributions

All authors contributed equally and have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

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ABBREVIATIONS

DHPS, deoxyhypusine synthase; eIFSA, eukaryotic translation initiation factor 5A; DOHH, deoxyhypusine hydroxylase; NAD, nicotinamide adenine dinucleotide; SAR, structure–activity relationship

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