Two-photon Doppler-free ultraviolet laser spectroscopy on sulphur atoms

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Abstract

The 3p4 3PJ–3p34p3PJ transition in the sulphur atom is investigated in a precision two-photon excitation scheme under Doppler-free and collision-free circumstances yielding an absolute accuracy of 0.0009 cm\(^{-1}\), using a narrowband pulsed laser. This verifies and improves the level separations between amply studied odd parity levels with even parity levels in S I. An improved value for the 3P2–3P1 ground state fine structure splitting is determined at 396.0564 (7) cm\(^{-1}\). A 34S–32S atomic isotope shift was measured from combining time-of-flight mass spectrometry with laser spectroscopy.

Keywords: sulphur atom, laser spectroscopy, isotope shift

(Some figures may appear in colour only in the online journal)

1. Introduction

Odd parity-level energies for neutral sulfur atoms have been extensively studied through VUV absorption spectroscopy from the ground electronic configuration [1–4]. The connection with even parity excited states is studied through visible and infrared spectroscopy involving transitions between excited states [5–8]. In addition, direct measurements of transitions between even parity states are studied through 2+1 resonance enhanced multiphoton ionization (REMPI) spectroscopy [9, 10].

The level energies of the 1D2 and 1S0 states of the 3p\(^4\) ground electronic configuration, were investigated via electric dipole-forbidden transitions, first measured by McConkey et al [11], and revisited with higher accuracy by Eriksson [12]. Based on the combination differences between the forbidden transitions, 3P1–1S0, 1D2–1S0 and 3P2–1D2 the level energies of the lowest five levels were determined at an uncertainty of 0.005 cm\(^{-1}\) [12]. Later, Brown et al measured the fine structure transition 3P1–3P0 using laser magnetic resonance yielding an accuracy better than 10\(^{-4}\) cm\(^{-1}\) [13]. The resulting level structure of the sulphur atom including a comprehensive compilation of lines and level energies is now well documented [14–16].

In the present study, high-resolution spectra of 3p\(^4\)3P1–3p\(^3\)4p3PJ transitions of 32S are measured by using 2+1 REMPI employing a narrowband pulsed laser amplifier in a scheme with counter-propagating laser beams, thus allowing for Doppler-free spectroscopy at high resolution and high accuracy. The study is aimed at accurately bridging the large energy gap between the ground state and the manifold of excited states, which can be probed at high accuracy via infrared and visible spectroscopies. Via this means the measurement of a few transitions will allow for improving the accuracy of the entire level structure of the sulphur atom. Moreover it will be shown that isotope shifts can be resolved in such Doppler-free precision experiment.

2. Experiment

The experimental setup, schematically shown in figure 1, is similar to that used for the production and detection of vibrationally excited states in molecular hydrogen, also obtained from photolysis of H\(_2\)S [17–19]. Two ultraviolet (UV) pulsed laser systems are used to produce sulphur atoms and perform the precise two-photon spectroscopy. Sulphur atoms in the 3p\(^4\)3P1 ground state triplet are formed by UV-photodissociation of H\(_2\)S molecules, a well-studied photolysis process [20–23]. The first UV-laser pulse, inducing the dissociation, is obtained from a frequency-doubled pulsed dye
The two-photon transition is measured by a traveling-wave pulsed-dye-amplifier (PDA) system amplifying the output of a narrowband cw-ring dye laser. The amplification is realized in three consecutive dye cells, pumped with the same Nd-YAG pump laser also used to pump the PDL [24]. The output of the PDA at 616–621 nm is frequency-doubled in a KDP crystal to provide UV-pulses in the range 308–311 nm with \( \sim 4 \) ns pulse width. The frequency of the cw-seed light is calibrated against the standard of I\(_2\) saturated hyperfine lines combined with the transmission markers of a stabilized Fabry–Perot interferometer [25]. The chirp effect on the pulses, giving rise to an effective frequency offset between the pulsed output of the PDA and the cw-light is assessed via optical heterodyne measurements and analyzed via known techniques [24, 26].

The narrowband UV beam is then split and configured in a counter-propagating beam setup to induce the Doppler-free two-photon transitions. The angle mismatch of the counter-propagating beams is reduced based on Sagnac interference fringes [27].

The UV beams are focused onto a spot of size few tens of \( \mu \)m spatially overlapping a pulsed \( \text{H}_2\text{S} \) beam, in a low-density region of a skimmed and collimated pulsed effusive gas expansion. To avoid AC-Stark disturbances from the dissociation laser, the PDA spectroscopy laser is optically delayed by 10 ns with respect to the photolysis laser, such that there is no temporal overlap. The sulphur atom signal is generated via 2\( ^+\) REMPI, whereby ions are extracted through a mass-resolving time-of-flight (TOF) tube, detecting S\(^+\) ions. Ion optics are triggered at a delay of \( \sim 50 \) ns from the spectroscopy laser, so that the laser-excitation takes place in a zero DC field. The ion signal is amplified by a microchannel plate with a phosphor imaging screen with detection on a photomultiplier tube. Mass-selected spectra are recorded with a boxcar integrator probing only a narrow channel of the TOF-trace. The large amounts of SH\(^+\) and \( \text{H}_2\text{S}^+ \) signal in nearby mass channels, as well as S\(^+\) background signals from various dissociation/ionization channels are limiting factors on the signal-to-noise-ratio of the S-atom spectra. In case of spectral recording of measuring spectra of \( ^{34}\text{S} \) this is even more detrimental.

3. Results and interpretation

All of the seven two-photon allowed transitions between 3\( p^4 3P\) and 3\( p^4 4P \) were measured in the wavelength interval 308–311 nm. Figure 2 displays recordings of all observed lines under Doppler-broadened conditions. Note that the combination \( J = 0 \leftrightarrow 1 \) is forbidden by two-photon selection rules [28].

The spectra for the 3\( p^4 3P \rightarrow 3p^4 4P \) and 3\( p^4 3P \rightarrow 3p^4 4P \) lines, recorded under Doppler-free conditions, are shown in greater detail in figures 3 and 4. The width of the spectral lines, measured at the lowest power, is about 290 MHz (FWHM), only slightly larger than expected by assuming exact Fourier-transform limited laser pulses of Gaussian spectral profile. The power dependence (or AC-Stark effect) for the transition frequencies is studied by varying the PDA pulse energy as shown in the inset of the figures. Table 1 lists the transition frequencies, upon extrapolation to zero
Figure 4. Spectrum of $3p^4\,{}^3P_2 - 3p^34p\,{}^3P_1$ two-photon transition of S I; same details as figure 3.

Table 1. Measured frequencies for the two-photon transitions $3p^4\,{}^1P_1 - 3p^34p\,{}^1P_1$ of $^{32}$S, with uncertainties indicated in parentheses.

| Initial state | Excited state | Obs. (cm$^{-1}$) |
|---------------|---------------|-----------------|
| $^3P_2$       | $^3P_1$       | 64888.9317 (9)  |
| $^3P_0$       | $^3P_1$       | 64891.3536 (9)  |
| $^3P_2$       | $^3P_0$       | 64892.5494 (9)  |
| $^3P_1$       | $^3P_0$       | 64492.8751 (9)  |
| $^3P_2$       | $^3P_0$       | 64317.7561 (9)  |
| $^3P_0$       | $^3P_0$       | 64317.7561 (9)  |

Table 2. Error budget for the two-photon frequencies for the sulphur atom measured in the present study, except for the $3p^4\,{}^3P_2 - 3p^34p\,{}^3P_2$ line, where the uncertainty is larger.

| Contribution | Uncertainty ($\times 10^{-4}$ cm$^{-1}$) |
|--------------|-----------------------------------------|
| Line profile (fitting) | 2 |
| Statistics | 3 |
| AC-Stark extrapolation | 5 |
| Frequency calibration | 3 |
| Cw-pulse offset (chirp) | 6 |
| Residual Doppler | < 1 |
| DC-Stark effect | < 1 |
| Total | 9 |

field, as measured for the seven transitions with the boxcar gate set to $^{32}$S.

The sources of uncertainty are summarized in an error budget presented in table 2. A statistical analysis of the determination of the line centres gives an uncertainty of $3 \times 10^{-4}$ cm$^{-1}$, including averaging over multiple recordings. The AC-Stark effect is the dominant systematic effect in the present study. It causes a shift of line centres, accompanied by broadening, and due to the spatial variation of laser intensity over the laser focus, also resulting in an asymmetry of the line profile [18]. The asymmetry was addressed by fitting skewed Voigt profiles. Analysis of the line shape results in an additional contribution to the uncertainty of $2 \times 10^{-4}$ cm$^{-1}$. The AC-Stark shift is further analyzed by performing measurements over a range of pulse energies of 20–100 μJ with extrapolation of the centre frequency to zero energy.

This adds a contribution to the error budget of $5 \times 10^{-4}$ cm$^{-1}$. Further contributions are associated with the frequency chirp in the PDA-system and the absolute frequency calibration against $I_2$-hyperfine components, which were analyzed by established techniques [24, 29] and result in a contribution of $6 \times 10^{-4}$ cm$^{-1}$ for the frequency uncertainty. The absolute frequency calibration against $I_2$ hyperfine components involves uncertainty in the reference frequencies [25] and measurement of the FSR, amounting to $3 \times 10^{-4}$ cm$^{-1}$. For the latter two contributions multiplication by four, for the frequency doubling and the two-photon process, is included. The experiment is essentially Doppler-free, although small shifts of the frequency centre may be associated with a non-isotropic velocity distribution of the S-atoms, similar to the case of H$_2$ investigated [30]. For this reason the counterpropagating laser beams were aligned in a Sagnac interferometer [27] limiting this effect to below $1 \times 10^{-4}$ cm$^{-1}$. Excitation was performed in zero field, hence the DC-Stark effect is negligible on the scale of the present accuracy. Taking the contributions in quadrature leads to a total uncertainty of 0.0009 cm$^{-1}$ for the frequencies of the two-photon resonances for all observed transitions except for one. The uncertainty of $3p^4\,{}^3P_1 - 3p^34p\,{}^3P_2$ is estimated at 0.0023 cm$^{-1}$ with larger uncertainty from statistics, a long measurement trace to be covered for reaching an I$_2$ resonance, and problems encountered in AC-Stark extrapolation.

In a single case, for the strongest line $^3P_2 - ^3P_2$, a study was made of the resonance line in $^{34}$S from the 5% naturally abundant isotope in the sample. The spectrum, shown in figure 5, and recorded with a boxcar gate probing mass-34, displays the much lower signal-to-noise ratio, caused by a spurious signal on mass-34 of H$_2$S$^+$ ions. When only considering the statistical line fitting and relative calibration errors the isotope shift on the resonances amounts to 0.0179 (4) cm$^{-1}$. The spectrum of figure 5 shows that the spectral contributions of $^{32}$S and $^{34}$S are well separated, thus verifying that the listed entries for the transition frequencies in table 1 pertain to the main $^{32}$S isotope, and do not correspond to a mixture of isotopes.
Table 3. Least-square fitting for level energy of \( 3p^4 4p \) \( 3P_1 \) and \( 3p^4 3P_0 \) of \( ^32S \), with uncertainties relative to ground \( 3p^4 3P_2 \) state indicated in parentheses. A comparison is given with values given in reference [14]; for the quoted uncertainties see text. All values are given in cm\(^{-1}\).

| Level         | This work | Ref. [14] | Difference |
|---------------|-----------|-----------|------------|
| \( 3p^4 3P_2 \) | 0         | 0         | —          |
| \( 3p^4 3P_1 \) | 396.0570(11) | 396.055(5) | 0.0020     |
| \( 3p^4 3P_0 \) | 573.5953(9)   | 573.640(16) | —0.0447    |
| \( 3p^4 4p \) \( 3P_1 \) | 64 888.9319(9) | 64 888.964(25) | —0.0321 |
| \( 3p^4 4p \) \( 3P_0 \) | 64 891.3525(8) | 64 891.380(25) | —0.0335 |
| \( 3p^4 4p \) \( 3P_2 \) | 64 892.5503(7) | 64 892.582(25) | —0.0317 |

Table 4. Least-square fitting for level energy of ground electron configuration \( 3p^6 \) of \( ^32S \), with uncertainties indicated within parentheses. The transition from references [12, 13] and this work are included in fitting, and a comparison is made with the data compilation of reference [14]. All values are given in cm\(^{-1}\).

| Level         | This work | Ref. [14] |
|---------------|-----------|-----------|
| \( 3P_2 \)   | 0         | 0         |
| \( 3P_1 \)   | 396.0564(7) | 396.055(5) |
| \( 3P_0 \)   | 573.5956(7) | 573.640(16) |
| \( 1D_2 \)   | 9238.6097(23) | 9238.609(5) |
| \( 1S_0 \)   | 22 179.9548(22) | 22 179.954(5) |

The isotope shift for a transition can be separated into normal mass shift, specific mass shift and field shift contributions [31]. In view of the very small wave function amplitude of the \( p \) outer electron within the nuclear charge radius, the field shift (or finite size) contribution is negligible for the differential isotope effect. The Bohr shift or normal mass shift \( \Delta E_{\text{SMS}} = E_{\text{SMS}}(^{34}S) - E_{\text{SMS}}(^{32}S) \) is calculated to be 0.0650 cm\(^{-1}\). From the experimental isotope shift obtained here, a specific mass shift of \( \Delta E_{\text{SMS}} = -0.0471(4) \) cm\(^{-1}\) is extracted for the \( 3p^4 3P_2 \rightarrow 3p^4 4p \) \( 3P_2 \) transition. The \( \Delta E_{\text{SMS}} \) experimental value will be useful in validating \( ab \text{ initi}o \) calculations of electron correlations.

4. Discussion: level energies

The precise determination of transition frequencies can be cast into a least-squares analysis to determine level energies in both \( 3P \) states in the \( 3p^4 \) and \( 3p^4 4p \) configurations using the LOPT program [32]. First the internal consistency of the measurements can be tested by including only the presently obtained data set, as is done in table 3. The LOPT analysis provides a consistent set of level energies at an accuracy in most cases below 10\(^{-3}\) cm\(^{-1}\). More importantly the deduced ground state splitting between \( 3P_1 \) and \( 3P_0 \) levels is in full agreement (within 1\(\sigma \)) with the very accurate LMR-measurement [13], the most precise level splitting determined in the sulphur atom. This agreement provides proof that the uncertainties of the present study are not underestimated.

In table 3 a comparison is also made with the results from VUV spectroscopy, which are at the basis of the comprehensive published line and level lists [1, 14, 15]. In these compilations a single spectral line in the VUV is included (a transition to the \( 4S \) \( 3S_1 \) level) for which an uncertainty as low as 0.025 cm\(^{-1}\) is stated [1]. We adopt this value as the general uncertainty for the level energies for the excited states, although the uncertainty for the overall level structure might be somewhat larger. The increased precision on the \( 3P \) ground state level energies in the compilations, so better than the quoted 0.025 cm\(^{-1}\), in fact derive from the rather accurate measurement of the forbidden transitions by Eriksson [12]. The \( 3P_0 \) level was not accessed in the measurement of reference [12], hence its uncertainty relies on

VUV-data. Viewed in this context the deviations between present results and the VUV-compilation [1], as listed in table 3 both for ground state splitting and \( 3p^4 4p \) excitation energies are close to the expected uncertainties. This includes the consistent shift —0.032 cm\(^{-1}\) for all three levels in the \( 3p^4 4p \) \( 3P \) excited triplet. This finding is indicative for an overall systematic shift of all excited levels in the data compilation by 0.03 cm\(^{-1}\).

Finally an LOPT least-squares analysis can be performed to determine the level energies of the entire \( 3p^4 \) ground electronic configuration including \( 1D_2 \) and \( 1S_0 \) levels, based on the present study in combination with the high precision measurements of magnetic dipole transitions in references [12, 13]. Table 4 lists the fitted level energy values with individual uncertainties relative to ground state and comparison with the corresponding values listed in the sulphur atom data compilation [14]. This results in an improved level structure for \( 3p^4 \), in particular for the lowest fine structure splitting \( 3P_2 - 3P_1 \) which is determined at 396.0564 (7) cm\(^{-1}\), corresponding to a far-infrared wavelength of 25.24893 (4) \( \mu \)m.

5. Conclusion

In conclusion, seven transitions in the \( 3p^4 3P_1 - 3p^4 4p 3P_1 \) multiplet are measured by narrowband laser spectroscopy at an uncertainty of 0.0009 cm\(^{-1}\). For the first time a \( ^34S_2 - ^32S_2 \) isotope shift has been measured in atomic sulphur, from which a value for the specific mass-shift was derived, a measure for electron correlations in the atom. The accurate transition frequencies improve the level energies of the \( 3p^4 \) \( 3P \) ground electronic configuration by a factor of two. The \( 3p^4 4p \) \( 3P_1 \) excited state level energies are determined at an absolute accuracy of less than 0.001 cm\(^{-1}\). The present study provides an indication of an overall systematic shift for the excited level energies as listed in spectroscopic data compilations for the sulphur atom [1, 14, 15]. The precise measurement of even parity excited states may help optimizing the level energies of odd parity levels by future improved measurements between excited states in the infrared and visible regions, therewith using the \( 3p^4 4p \) \( 3P_1 \) levels as anchor levels, in a similar fashion as applied to H\(_2\) [33].

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References

[1] Kaufman V 1982 The spectrum of neutral sulphur (S I) in the vacuum ultra-violet Phys. Scr. 26 439
[2] Sarma V N and Joshi Y N 1984 New lines of atomic sulphur and atomic mercury observed in the vacuum ultraviolet region Physica B+C 123 349
[3] Joshi Y N, Mazzoni M, Nencioni A, Parkinson W H and Cantu A 1987 Photosorbsorption spectrum of atomic sulphur J. Phys. B: At. Mol. Phys. 20 1203
[4] Zhou J, Jones B, Yang X, Jackson W M and Ng C Y 2008 Zhou J et al 2020 Ultraviolet photolysis of H2S and its implications for SH radical production in the interstellar medium Nat. Commun. 11 1547
[5] Baclawski A and Musielok J 2011 Experimental and ab initio re-investigation of the Lyman-α photodissociation of H2S and D2S J. Chem. Phys. 114 1672
[6] Xu S, van Dierendonck R, Dixon R N and Ashfold M N R 2001 The spectrum of neutral sulphur from the infrared part of the vacuum ultraviolet Appl. Phys. B 123 2961
[7] Melikechi N, Gangopadhyay S and Eyler E E 1994 Phase dynamics in nanosecond pulsed dye-laser amplification J. Opt. Soc. Am. B 11 2469
[8] Kramida A E 2011 The program LOPT for least-squares optimization of energy levels Comput. Phys. Commun. 182 419
[9] Barlow D, Sulambides E, Vervoort M and Ubachs W 2010 Accurate level energies in the B^1Σ_u^+, C^1Π_u, B^1Σ_g^+, 1^1Π_g, 1^1Δ_g states of H2 Mol. Phys. 108 827
[10] Carette T, Drag C, Scharf O, Blondel C, Delsart C, Froese Fischer C and Godefroid M 2010 Isotope shift in the sulphur photoabsorption spectrum of atomicsulphur for lifetime measurements and precision spectroscopy J. Opt. Soc. Am. B 27 2469
[11] Cheng C-F et al 2018 Dissociation energy of the hydrogen molecule at 10–9 accuracy Phys. Rev. Lett. 121 013001
[12] Carette T, Drag C, Scharf O, Blondel C, Delsart C, Froese Fischer C and Godefroid M 2010 Isotope shift in the sulphur electron affinity: observation and theory Phys. Rev. A 81 042522
[13] Kramida A E 2011 The program LOPT for least-squares optimization of energy levels Comput. Phys. Commun. 182 419
[14] Barlow D, Sulambides E, Vervoort M and Ubachs W 2010 Accurate level energies in the B^1Σ_u^+, C^1Π_u, B^1Σ_g^+, 1^1Π_g, 1^1Δ_g states of H2 Mol. Phys. 108 827
[15] Kaufman V and Martin W C 1993 Vibrational and energy level classifications for the spectra of sulphur (S I) through (S XVI) J. Phys. Chem. Ref. Data 22 279
[16] Morton D C 2003 Atomic data for resonance absorption lines III. Wavelengths longward of the Lyman limit for the elements hydrogen to gallium Astrophys. J. Suppl. Ser. 149 205
[17] Kaufman V, Sulambides E J and Ubachs W 2015 Communications: test of quantum chemistry in vibrationally hot hydrogen molecules J. Chem. Phys. 143 081102
[18] Trivikram T M, Niu M L, Wcislo P, Ubachs W and Sulambides E J 2016 Precision measurements and test of molecular theory in highly excited vibrational states of H2 (v = 11) Appl. Phys. B 122 294
[19] Steadman J and Baer T 1988 The production and spectroscopy of excited sulphur atoms from the two-photon dissociation of H2S J. Chem. Phys. 89 5507
[20] Steadman J and Baer T 1989 The production and characterization by resonance enhanced multiphoton ionization of H2 (v = 10 – 14) from photodissociation of H2S J. Chem. Phys. 91 6113
[21] Cook P A, Langford S R, Dixon R N and Ashfold M N R 2001 An experimental and ab initio re-investigation of the Lyman-α photodissociation of H2S and D2S J. Chem. Phys. 114 1672
[22] Zhou J et al 2020 Ultraviolet photolysis of H2S and its implications for SH radical production in the interstellar medium Nat. Commun. 11 1547
[23] Eikema K S E, Ubachs W, Vassen W and Hogervorst W 1997 Lamb shift measurement in the 1^1S ground state of helium Phys. Rev. A 55 1866
[24] Steadman J and Baer T 1989 The production and characterization by resonance enhanced multiphoton ionization of H2 (v = 10 – 14) from photodissociation of H2S J. Chem. Phys. 91 6113
[25] Cook P A, Langford S R, Dixon R N and Ashfold M N R 2001 An experimental and ab initio re-investigation of the Lyman-α photodissociation of H2S and D2S J. Chem. Phys. 114 1672
[26] Zhou J et al 2020 Ultraviolet photolysis of H2S and its implications for SH radical production in the interstellar medium Nat. Commun. 11 1547
[27] Eikema K S E, Ubachs W, Vassen W and Hogervorst W 1997 Lamb shift measurement in the 1^1S ground state of helium Phys. Rev. A 55 1866
[28] Steadman J and Baer T 1989 The production and characterization by resonance enhanced multiphoton ionization of H2 (v = 10 – 14) from photodissociation of H2S J. Chem. Phys. 91 6113
[29] Melikechi N, Gangopadhyay S and Eyler E E 1994 Phase dynamics in nanosecond pulsed dye-laser amplification J. Opt. Soc. Am. B 11 2469
[30] Kramida A E 2011 The program LOPT for least-squares optimization of energy levels Comput. Phys. Commun. 182 419
[31] Barlow D, Sulambides E, Vervoort M and Ubachs W 2010 Accurate level energies in the B^1Σ_u^+, C^1Π_u, B^1Σ_g^+, 1^1Π_g, 1^1Δ_g states of H2 Mol. Phys. 108 827
[32] Kramida A E 2011 The program LOPT for least-squares optimization of energy levels Comput. Phys. Commun. 182 419
[33] Barlow D, Sulambides E, Vervoort M and Ubachs W 2010 Accurate level energies in the B^1Σ_u^+, C^1Π_u, B^1Σ_g^+, 1^1Π_g, 1^1Δ_g states of H2 Mol. Phys. 108 827