Association between Malondialdehyde and Glutathione (L-gamma-Glutamyl-Cysteinyl-Glycine/GSH) Levels on Workers Exposed to Benzene in Indonesia

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Abstract

BACKGROUND: Chemicals that enter the body, especially benzene, will undergo a detoxification process. Unfortunately, at the detoxification process, sometimes benzene can produce free radicals. Free radical oxidation of lipids produces MDA compounds (malondialdehyde). To overcome these free radicals, the body will adapt to produce Glutathione (GSH) enzymes.

AIM: The purpose of this study was to analyse the relationship between benzene concentration, MDA levels and glutathione enzymes in Shoe-Maker Home Industry workers exposed to benzene for more than 10 years.

METHODS: Measurement of benzene concentration using a gas chromatography-flame ionisation detector (GC-FID). MDA levels used a modified spectrophotometric and GSH method of thiobarbituric acid (TBA) test.

RESULT: The results showed that the majority of respondents had benzene concentrations still below the TLV value, mean of MDA levels were 6.94 mg/ml, while GSH was 4.54 mg/ml. Benzene concentration did not have a significant correlation with MDA and glutathione levels, whereas MDA levels had a strong correlation with glutathione levels (p = 0.000; r = 0.947).

CONCLUSION: Workers should always use PPE and always eat foods that contain lots of glutathione enzymes such as spinach or broccoli to reduce the impact of free radicals from benzene inhalation.

Introduction

Benzene is a liquid that is colourless and has a sweet smell, evaporates very rapidly in the air, and is difficult to dissolve in water [1], [2]. Benzene is also a raw material for making plastics, resins, synthetic fibres, dyes, detergents, medicines, pesticides, and components of crude oil, gasoline, and cigarette smoke [3], [4]. Pathways to benzene exposure can be through the skin, respiratory tract, mouth and then to the digestive tract [5].

A person who is exposed to high levels of benzene can experience several signs and symptoms, including drowsiness, dizziness, rapid or irregular heartbeat, headaches, tremors, confusion, unconsciousness, until death [6]. Excessive benzene in the body can become free radicals that can reduce blood cell production [7].

Free radicals are compounds or atoms that have one or more unpaired electrons [8]. Free radicals oxidise some of the body macromolecules such as proteins, nucleic acids and lipids [9]. Free radical oxidation of proteins, nucleic acids, and lipids each produces carbonyl compounds, MDA (malondialdehyde) and deoxyguanosine P [10].

To overcome free radicals, the body needs
antioxidants. Antioxidants are obtained from outside the body (food) or produced by the body itself (endogenous) [11]. Examples of endogenous antioxidants are superoxide dismutase, glutathione (GSH), catalase and glutathione peroxidase. One antioxidant that is often measured to see the impact of an increase in free radicals in the body is GSH [12, 13]. The Glutathione molecular formula is C10H17N3O6S, with a molecular weight of 307.3235 g/mol. As an antioxidant for the body, glutathione is a tripeptide consisting of amino acids; glutamate, cysteine, and glycine [14]. The content of glutathione is found in most of the body's cells, but the most are in the liver. The thiol (SH) group of cysteine functions as a proton donor and is responsible for the biological activity of glutathione. Suggested food sources because they contain glutathione as antioxidants namely asparagus, spinach, broccoli, garlic, kale, onions, watercress, cabbage, Brussels sprouts, some herbs like turmeric, cinnamon, watermelon, avocado, grapes, peaches, oranges, walnuts, granola, turkey and chicken meat, cottage cheese and yoghurt [13].

Several studies have shown that benzene exposure can increase free radicals and reduce the body's antioxidant status, especially GSH enzymes. Research in Jakarta and Iraq shows that gas station workers exposed to benzene are more susceptible to DNA damage due to free radicals. Research on the relationship of benzene, MDA and glutathione concentrations in shoe Home Industry workers still does not exist in Indonesia.

Therefore, this study aims to analyse the relationship of benzene concentration, MDA levels and glutathione enzymes in Home Industry workers exposed to benzene.

Material and Methods

The type of research used is cross-sectional. Subjects are workers in the Tambak Osowilangun shoe industry in Surabaya. The inclusion criteria in this study were male and female workers who had worked in the shoe industry in Tambak Osowilangun for > 10 years and were willing to be used as research respondents. The study sample was 25 people.

The variables calculated were benzene levels, MDA levels, GSH levels, and measurements of benzene concentration at five points in the industry. The research subject was chosen after the person was willing to participate in the study by first describing the benefits and inconvenience of participating in the study. Willingness to participate in the study was made in writing through informed consent, and this study had received prior ethical approval by the Public Health Faculty Ethics Committee, Airlangga University with ethics number 516 KEP-K.

Measurement of length of work, average work every day, and work time in a week are obtained through in-depth interviews with respondents. Then, the measurement of benzene concentration in the work environment using a measurement method of NIOSH 1501 with an activated carbon (charcoal) pipe which uses a gas chromatography-flame ionisation detector (GC-FID) technique using NIOSH 1501 standard.

MDA measurements were carried out using a modified spectrophotometry method of thiobarbituric acid (TBA) test. A total of 400 µl of the sample was reacted with 200 µl of trichloroacetic acid (TCA) 20% for deproteination. Then the cortex and centrifuge at a speed of 5000 rpm for 10 minutes. The supernatant formed was taken and 400 µl TBA 0.67% was added. Then the sample was vortexed and incubated in a water heater at 96°C, 10 minutes then lift and cool at room temperature. Then read the absorption at a wavelength of 530 nm. The sample is taken immediately after the employee's work shift is finished [14]. The normal average level for MDA is 2.61 µmol/L [15].

GSH measurement with a sample of blood taken from the left arm cubital vein as much as 2 mL of research subjects. The blood is then centrifuged at 2000 rpm for 3 minutes to get a plasma. Then the plasma is stored in a refrigerator -20°C, before GSH examination. GSH levels were measured by mixing 50 plasma with 1.78 mL phosphate buffer 0.1 M pH 8 and 0.2-mL TCA 5%. The mixture was then centrifuged at 1500 g for 5 minutes, a temperature of 40°C. The supernatant was then added with 0.01 mL DNTB and left for 1 hour. The mixture is then examined using spectrophotometry at a wavelength of 412 nm to determine plasma GSH levels [16]. GSH normal level is 3.8-5.5 µmol/L [17].

Statistical analysis using Pearson and Spearman's Rank correlation test with an alpha of 0.05. The closer to 1, the stronger the correlation between variables and vice versa.

Results

Characteristics of Shoe Maker Worker

The majority of respondents have a high school senior education background (45.80%) and male sex (56%) and do not have smoking habits (60%). The majority of respondents, as many as 56% still have benzene concentrations still under TLV standard (0.01 mg/ml). In Figure 1, shows that the average MDA level in the respondent's body is 6.94 mg/ml.
The highest MDA level in the respondent's body was 12.73 mg/ml, while the low level was 2.93 mg/ml. Figure 1 also shows that the average level of glutathione in the body of the respondent was 4.54 mg/ml. The highest level of glutathione in the body of the respondent was 8.28 mg/ml, while the lowest level was 1.86 mg/ml.

Correlation between Benzene, MDA and Glutathione Concentration in Shoe Home Industry Worker

Correlation between Benzene concentration and MDA

Figure 2 shows that there is no trend curve between the relationship between benzene concentration and MDA level. This is by the statistical test which states that there is no correlation between benzene concentration and MDA level \((p = 1.000; r = 0.000)\). Correlation between Benzene Concentration with Glutathione Level

Figure 3 shows that there is no tendency curve between the relationship of benzene concentration and glutathione level. This is by the statistical test which states that there is no relationship between benzene concentration and glutathione level \((p = 1.000; r = 0.000)\).

Discussion

The majority of respondents have male gender and have a high school education background (high school). Benzene concentration in the majority of respondents has a value under the Threshold Limit Value (TLV). The Threshold value for the concentration of benzene according to the Regulation of the Minister of Manpower and Transmigration of the Republic of Indonesia Number 13 of 2011 concerning the Physics and Chemical Factor Threshold Value at the Workplace is 0.5 ppm (1.6 mg/m³) [18].

Malondialdehyde (MDA) is a compound that can describe the activity of free radicals in cells so that it is used as one of the indications of oxidative
stress caused by free radicals [19]. Another study reinforces this statement by stating that the mediator Malondialdehyde (MDA) is a final product of fat peroxidation which is used as a biological biomarker of fat peroxidation and can describe the degree of oxidative stress [20]. The average MDA level in the respondent's body was 6.94 mg/ml, and this value showed a higher value than previous studies in Indonesia (0.731 nmol/mL). Statistical tests also showed no significant relationship between benzene and MDA levels in shoe workers. This contrasts with similar research [10]. But the thing that distinguishes this research from the previous one is the location/place of research. The previous research location was located at a gas station which was known to have higher levels of benzene than the shoe industry factory which was only exposed to benzene in the glueing process. Other factors that influence can be too small a large sample and less exposure to benzene.

Glutathione (γ-glutamyl-cysteinylglycine) is a tripeptide consisting of glutamic acid, cysteine, and glycine. The compound has a sulfhydryl/thiol group (-SH) found in the amino acid cysteine. The sulfhydryl group causes GSH to act as a powerful electron donor (nucleophile) in counteracting free radicals. GSH can decompose H₂O₂ to H₂O with the help of glutathione peroxidase enzymes [12], [21]. These compounds are metabolised by the liver to produce free radicals such as superoxide anions (O²⁻), hydroxyl radicals (OH⁻) and semicunionan radicals. GSH works to counteract these free radicals to prevent or reduce cell damage [22]. The longer you work at the benzene exposure site, the more exposure to benzene, toluene and xylene compounds will accumulate and are more likely to reduce the body's antioxidants [11].

GSH has a role as an antioxidant by reducing free radicals directly or as a cofactor of antioxidant enzymes such as glutathione peroxidase (Figure 1) and glutathione transhydrogenase [23]. The main function of GSH is to detoxify drugs, xenobiotics or pesticides catalysed by GSH-S-transferase enzymes. GSH also plays a role in maintaining thiol groups (-SH) in essential proteins, by reducing disulfide bonds in proteins, which are catalysed by the enzyme thioltransferase [12], [23]. However, the results of the statistical test analysis stated that there was no significant relationship between benzene concentration and glutathione. This has similarities with previous studies [16]. The thing that can be a significant factor of this correlation is the possibility that there are many other antioxidants (glutathione peroxidase, SOD and catalase) which play a role in reducing free radicals [8], [12].

MDA and glutathione levels have a strong significant correlation and have a reciprocal relationship which means that the higher the MDA level, the lower the level of glutathione in the worker's body. This has similarities with another study which states that there is an increase in MDA levels and a decrease in Glutathione levels in cement workers [24]. Even studies in Jordan and Poland in automobile workers showed an increase in MDA levels by 48% and a decrease in GSH levels by 16-25% [25], [26]. High MDA levels indicate higher free radicals as well. This benzene radical can suppress detoxification enzymes, one of which is GSH. The most dangerous forms of free radicals benzene are superoxide (O₂⁻) anion, hydroxyl radical (OH) and hypochlorite acid (HOCl) and hydrogen peroxide (H₂O₂). Also, benzene free radicals can damage blood cells, can cause lipid peroxidation which can cause liver fibrosis [27]. What can be done to reduce benzene exposure in workers is to use PPE regularly such as masks, especially during the glueing process. Another approach is taken by consuming foods rich in detoxification enzymes, especially GSH which is usually found in the majority of vegetables such as asparagus, spinach, broccoli, garlic, kale, and onions [13].

In conclusion, the majority of respondents have male gender, high school education background, benzene concentration is still below the TLV value, the average MDA level is 6.94 mg/ml, and the average glutathione level is 4.54 mg/ml. Benzene concentration did not have a significant correlation with MDA and glutathione levels; on the contrary, MDA levels had a strong correlation with glutathione levels. The influencing factor was the benzene threshold which was still too small, so it still did not show high levels of GSH and MDA. Conversely, MDA and GSH have a strong correlation because the more benzene that becomes free radicals (MDA), the free radicals will directly interfere with the work of biotransformation enzymes (detoxification), one of which is GSH. Workers should always use PPE and always eat foods that contain lots of glutathione.

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**Figure 5: Role of GSH to Prevent Free Radicals**

1. NADPH Oxidase; 2. Superoxide Dismutase; 3. Glutathione Peroxidase; 4. Glutathione S-Transferase; 5. Glutathione Reductase
enzymes such as vegetables (spinach or broccoli) to reduce the impact of free radicals from benzene inhalation.

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