Pressure-Induced Phase Transformation in $\beta$-Eucryptite: an X-Ray Diffraction and Density Functional Theory Study

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Certain aluminosilicates display exotic properties enabled by their framework structure made of corner-sharing tetrahedral rigid units. Using in situ diamond-anvil cell x-ray diffraction (XRD), we study the pressure-induced transformation of $\beta$ eucryptite, a prototypical aluminosilicate. $\beta$ eucryptite undergoes a phase transformation at moderate pressures, but the atomic structure of the new phase has not yet been reported. Based on density functional theory stability studies and Rietveld analysis of XRD patterns, we find that the pressure-stabilized phase belongs to the Pna$_2$$^1$ space group. Furthermore, we discover two other possible pressure-stabilized polymorphs, P1c1 and Pca$_2$$^1$.

From their ubiquity in silica phases found in Earth’s mantle$^{1–3}$ to their importance for synthesizing ceramics with extraordinary electronic$^{4,5}$ or mechanical properties$^{6,7}$, pressure-induced phase transformations in silicates provide fundamental insights into structure–property relations as well as ways to control these relations so as to enable a variety of engineering applications. Such applications range from transformation toughening in ceramics$^7$ to cathode materials for Li-ion batteries.$^{8,9}$ Amongst silicates, LiAlSiO$_4$ displays a wide variety of pressure-induced transformations. Their structure consists of tetrahedra centered at Si or Al, connected in an anisotropic framework that weakly binds lithium ions in the voids surrounded by the tetrahedra.$^{10}$ This framework underlies a whole host of exotic phenomena and properties, such as negative coefficient of thermal expansion$^{11–15}$, negative compressibility, and one-dimensional ionic conduction.$^{13,16–19}$ In addition, $\beta$-eucryptite (a LiAlSiO$_4$ polymorph) exhibits a pressure-induced phase transformation at pressures that are low enough to be exploited technologically.$^{20,21}$

The crystal structure of $\beta$-eucryptite (space group P6$_3$$^1$22 or P6$_3$$^2$22) can be described as a stuffed derivative of $\beta$-quartz$^{10,12,14}$.$^*$ For this structure, Morosin et al.$^{15}$ reported a phase transformation occurring at pressures as low as ~0.8 GPa. Zhang et al.$^{22,23}$ also found a new polymorph (dubbed the $\epsilon$ phase) around 1 GPa, reporting that the transformation from $\beta$ to $\epsilon$ was reversible; amorphization was found to occur at pressures above 5 GPa.$^21$ More recent indentation and Raman spectroscopy experiments revealed that the critical pressure for the forward transformation ($\beta$ to $\epsilon$) is higher than that for the reverse one, suggesting that it may be possible to produce metastable eucryptite phases under ambient conditions.

While there is no doubt that a phase transformation from $\beta$-eucryptite to another crystalline phase exists at low pressures, the atomic structure of the $\epsilon$ phase remains unknown. To date, the crystal system, lattice constants, and Miller indices associated with the $\epsilon$ phase diffraction peaks reported by Zhang et al.$^{22}$ have not been reproduced by other groups. Here, we have carried out x-ray diffraction (XRD) experiments and density functional theory (DFT) calculations in order to elucidate the atomic structure of the $\epsilon$ phase. We have found that there are several polymorphs that become more stable than the $\beta$ phase upon increasing pressure. From these, we have determined that the best match for the experimental XRD patterns is achieved by an orthorhombic Pna$_2$$^1$ structure. Furthermore, we have found that the presence of Mg as a dopant leads to the observation of two phases that coexist past the transition pressure, and identified their atomic structures as well. This finding indicates a subtle competition between polymorphs under pressure, and suggests a way to modify the number and type of coexisting phases that could be used in transformation toughening.

In our experiments, $\beta$-eucryptite powders have been synthesized through a sol-gel route.$^{24,25}$ We have doped some of the samples with Mg, substituting for lithium up to 0.3 mol%. XRD measurements under pressure have been performed in situ at room temperature in a diamond anvil cell (DAC), from ambient pressure to 5 GPa; more details are given in Supplemental Material (SM). The pressure $p$ is derived from changes in the ruby fluorescence$^{26}$ via $p$(GPa)$=0.274\Delta A$, where $\Delta A$(A) is the difference between the ruby wavelength being detected and that at ambient pressure, 6942.1 A.$^{27}$ As evidenced by the lack of band splitting in ruby fluorescence, shear was absent$^{23,24}$ in the sample, so all experiments have been carried out in hydrostatic conditions.

In situ DAC XRD experiments were conducted using an angle dispersive synchrotron source at Cornell High Energy Synchrotron Source (CHESS).$^{26}$ Incident X-rays were converted by a double-bouncing monochromator at
FIG. 1: XRD patterns showing the pressure-induced phase transformation in \( \beta \)-eucryptite. (a) Variation of the pattern with pressure in a loading-unloading cycle. The patterns that contain the new phase(s) are shown in black, while the \( \beta \)-phase is red. (b)-(d) Detail views of the main peak at three pressures around the transition, showing the main peaks (b) before, (c) during, and (d) after the transition.

a wavelength of \( \lambda = 0.48596 \text{Å} \). Our data [Fig. 1(a)] shows that at ambient pressure only the \( \beta \) phase is present, which has a dominant peak [Fig. 1(b)] corresponding to the (202) planes. Above a threshold pressure, this peak changes shape as shown in Fig. 1(c): new, smaller peaks appear at that pressure, and are marked by black dots on the 2.47GPa curve in Fig. 1(a). By fitting the complex peak shape at 2.47GPa [Fig. 1(c)] with Lorentzian functions, we identify three peak locations –one corresponding to \( \beta \), and the other two corresponding to the new phase(s). A small increase in pressure past 2.47 GPa leads to the disappearance of the main \( \beta \) peak, while the intensity of the other two peaks increases [Fig. 1(d)]. Thus, for the Mg-doped samples analyzed, the transition occurs between 2.47GPa and 2.83GPa, with the \( \beta \) phase absent past 2.83GPa; pure samples exhibit somewhat smaller transition pressures.

We now focus on searching for crystalline phases that can become more stable than \( \beta \) upon pressure loading, and hence could be the product of the pressure-induced phase transformation of \( \beta \). To this end, we have analyzed over thirty structures made of corner-sharing AlO\(_4\) and SiO\(_4\) tetrahedra such that any Al-centered tetrahedron is surrounded by four Si-centered tetrahedra, and vice-versa. These structures have been obtained in two ways: (a) by creating Li-stuffed derivatives of known SiO\(_2\) structures, similar to the way in which \( \beta \)-cryptbite is related to \( \beta \)-quartz; and (b) by converting the alkaline atom M (M = Li, Na, K) of known MAlSiO\(_4\) and MAlGeO\(_4\) structures\(^{35}\) into Li, and Ge into Si when necessary. All structures have been relaxed using DFT in the generalized gradient approximation (GGA)\(^{33}\) for each pressure from 0 to 5 GPa in increments of 0.5 GPa. After eliminating duplicates, the crystals that we identified as having higher density than \( \beta \) and low heat of formation are shown in Fig. 2, with the space group and density at zero pressure identified; \( \beta \) is included as reference. The higher-density phases also have lower symmetry, as expected from the symmetry hierarchy of crystal systems.\(^{37}\) Two of the three new phases are orthorhombic, with space groups no. 33 (Pna\(_2\)1\(_1\)) and 29 (Pca\(_2\)1\(_1\)); the remaining phase is monoclinic (group no. 7, P1c\(_1\)).

Next, we discuss the relative stability of these phases with respect to \( \beta \). In Fig. 3, we plot the difference between the enthalpy of each candidate phase and that of \( \beta \), per unit formula. Interestingly, past a certain pressure each of these phases becomes more stable than \( \beta \) eucryptite (Fig. 3); for these pressure-stabilized phases, the lattice parameters are given in Table I, and the atomic positions are given in SM. Furthermore, all threshold pressures fall between 2 GPa and 3 GPa (Fig. 3), in good agreement with the experimental assessment of the range of pressures required for transition (Fig. 1). We cannot expect a more quantitative agreement because of the uncertainties related to the indirect pressure measurements in the DAC, and of the approximations made in the DFT calculations (for example, the enthalpies are computed at 0 K and not at room temperature). In order to identify more precisely the phases to which \( \beta \) transitions upon loading, we examine the XRD patterns at high-pressures.
FIG. 2: Crystal structures of (a) \( \beta \)-eucryptite and (b)-(d) three denser polymorphs, and their densities at zero pressure. Al-(Si-)centered tetrahedra are shown in gray (tan), Li atoms are purple spheres.

FIG. 3: Pressure-dependent enthalpy of LiAlSiO\(_4\) polymorphs, with respect to the \( \beta \)-phase.

We have performed Rietveld analysis of the XRD data using MAUD software\(^{38}\) designed to refine the background, structural (atomic coordinates, occupancies, lattice parameters and angles), and microstructural (particle size, lattice strain, residual stress, texture, etc) parameters via a least-squares method\(^{39}\). As input, we have supplied the XRD pattern and a candidate structure with its atomic coordinates and symmetry group. The optimization proceeds with fitting the background and the peak shape; the shape of the peaks was assumed to be of asymmetric, pseudo-Voigt form so it can fit crystal size and strain broadening of the experimental profiles\(^{40}\).

Without refining the fractional atomic coordinates and thermal vibration parameters, the following parameters have been optimized: background coefficients, scale, lattice parameters, zero-shift error, and peak shape parameters. The procedure yields a simulated optimized spectrum for comparison with the experimental data. We have carried out this procedure for each of the candidate structures [Fig. 2(b)-(d)]. The best fit for the XRD pattern of the pure, undoped sample is the Pna\(_2\)\(_1\) structure, shown in Fig. 4(a) for an XRD pattern at \( \sim 5 \) GPa.

It is worthwhile to compare this result with the previous reports on the phase transformations of \( \beta \) eucryptite. Morosin et al. observed a phase transformation around 0.8 GPa, and interpreted it as a hexagonal phase with lattice constants commensurate with half of those of \( \beta \)\(^{15}\). The structure of our monoclinic P1c1 phase is stabilized at the lowest pressure (Fig. 3) and is somewhat close to being hexagonal (\( a \) and \( c \) are within 1.1% of each other, and the angle between them \( \sim 12\% \) away from 120\(^\circ\)), so
TABLE I: Lattice parameters and angles for the crystal structures shown in Fig. 2, computed at zero pressure using GGA. The dependence on pressure is included in SM.

| Phase   | a (Å) | b (Å) | c (Å) | α (°) | β (°) | γ (°) |
|---------|-------|-------|-------|-------|-------|-------|
| P1c1    | 10.575| 10.575| 11.391| 90.000| 90.000| 120.000|
| Pca2₁   | 8.200 | 8.288 | 5.147 | 90.000| 90.000| 106.290|
| Pna2₁   | 10.079| 5.033 | 13.114| 90.000| 90.000| 90.000|
| Pna2₂   | 10.082| 6.673 | 4.990 | 90.000| 90.000| 90.000|

10.575 10.575 11.391 90.000 90.000 120.000
10.082 6.673 4.990 90.000 90.000 90.000
10.575 10.575 11.391 90.000 90.000 120.000

It is reasonable to infer that this phase may actually be the one found by Morosin et al. Zhao et al. reported a reversible transformation around 1.5 GPa to ε eucryptite, although these authors conclude that the ε phase is orthorhombic, its space group and the atomic structure have not been reported so far. Our data also indicates that ε eucryptite is orthorhombic, but with different lattice constants; in addition, we report here the space group (no. 33, Pna2₁) and the atomic structure [Fig. 2(e) and SM]. The previous indexing is not consistent with the reflection conditions for most orthorhombic groups. On the other hand, the planes that we have found in our XRD pattern [indexed in Fig. 2(a) and SM] obey the reflection conditions for group no. 33, and are largely the same as those encountered in other materials with the space group Pna2₁.

We have also analysed XRD data of 0.3% Mg doped samples such as those in Fig. 1 and have found that no single phase results in a satisfactory fit to the XRD data. Therefore, we are led to assume that in the presence of Mg, there can be two or more phases present at high pressures. We have analyzed phase mixtures using MAUD for all possible combinations of the phases shown in Fig. 2(b)-(d). The best fit for the XRD pattern of the Mg-doped eucryptite at 5 GPa [shown in Fig. 2(b)] is obtained from a mixture of 45.3% (by volume) Pna2₁ and 54.7% Pca2₁. It is rather intriguing that the presence of Mg in very small amounts facilitates the occurrence of more than one crystalline phase upon pressure loading—which may be due to entropic (mixing) or kinetic factors.

In summary, we have used XRD measurements and DFT calculations to elucidate the structure of the pressure-stabilized ε phase of eucryptite. While this answers a long-standing question, deeper investigations are necessary to understand how to control the presence of different polymorphs at low and moderate pressures. The reversible nature $\beta \rightarrow \epsilon$ transition, coupled with moderate values of the transition pressure (2-3 GPa) and of volume change (~9%), make it suitable for transformation toughening of ceramic composites for various applications. Knowledge of the ε phase can enable a better control over the design of such composites (e.g., optimizing the particle size, and distribution of ε-eucryptite particles to be employed in a suitable matrix). Furthermore, we have found that the presence of small amounts of dopants facilitates the coexistence of distinct polymorphs under pressure, which may lead to novel properties displayed by such phase mixtures.

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