Research Article

Workability and Compressive Strength Behavior of a Cemented High-Porosity Backfill Material

W. H. Cao,1,2 X. F. Wang,1,3,4 D. S. Zhang,1 X. J. Ji,2 X. Z. Chen,2 A. Zhang,2 and F. X. Zhu2

1State Key Laboratory of Coal Resources and Safe Mining, China University of Mining & Technology, Xuzhou, China
2School of Civil Engineering, Nanyang Institute of Technology, Nanyang, China
3Laboratory of Mine Earthquake Monitoring and Prevention, China University of Mining & Technology, Xuzhou, China
4School of Mines, China University of Mining & Technology, Xuzhou, China

Correspondence should be addressed to X. F. Wang; wangxufeng@cumt.edu.cn

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1.Introduction

Underground mining leads to a series of mining damage and environmental problems, such as the destruction of aquifer resources, surface subsidence damage to ground buildings, and surface vegetation death caused by ecological environmental damage [1, 2], which will become more prominent, with the development of coal resources gradually shifting its focus to arid and semiarid areas in western China [3]. As the most effective method to control surface subsidence and realize green mining, the filling and mining technology plays an essential role in the process of coal resource mining for its unique mining technique [4].

Filling material is the decisive factor and the bottleneck of the development of the filling mining. Over the past few decades, scholars have been looking for filling materials with sufficient raw materials, low cost, stable mechanical properties, and good fluidity. Due to the large amount of solid wastes such as waste rock, coal gangue, and tailings, the reasonable use of solid wastes to prepare composite cemented filling materials not only reduces the environmental pollution and surface waste treatment costs but also effectively reduces the cost of filling mining [5–8].

At present, with the development of cemented filling technology, remarkable achievements have been made in the preparation and properties of different solid waste cement filling materials. The particle size of gangue has a significant effect on the strength of CRF, which is also known as consolidated rockfill [9]. Cemented paste backfill (CPB) and cemented tailings backfill (CTB) are widely used in underground metal mining [10]; fly ash, blast furnace slag (slag), silica fume, pozzolanic ash, and alkali-activated materials were used as binder instead of ordinary Portland cement to reduce the cost of CPB mix [11–14]. At the same time, the properties and ratio of different types of new cementing materials, such as soil paste filling material [15],
gangue-waste concrete cemented paste backfill [16], and coal gangue backfill [17], are studied. These studies have greatly enriched the content of cemented filling materials in underground mine.

However, the main raw materials of cement paste materials, such as coal gangue, tailings, and other solid wastes, are still far less than the mining space, so it is difficult to meet the needs of filling in large-volume goaf, especially in the urgent need of green mining in mining areas of western China.

It is exciting that the western mining area is rich in surface sand. The author uses the surface sand of the mining area as the main raw material, mixes it with cementitious material into slurry, and adds foam to form the cemented high-porosity backfill (CHPB) material, which has the advantages of light weight and low cost [18]. The fresh (workability index) and hardened (unconfined compressive strength-UCS) are two main factors when designing an effective, lucrative, and sustainable CHPB matrix for underground structures [19, 20]. The main applications of CHPB in the underground mining industry are roof-contacted filling [21], small pit goaf filling [22], fire prevention, and plugging air leakage in goaf [23]. The effects of the raw material ratio and foam expansion rate on curing time and compressive strength of CHPB were studied [24], and the UCS prediction model for foam-cemented paste backfill was proposed. These studies have important reference to the research of this paper.

The objectives of this study are to therefore experimentally investigate the impacts of a number of parameters such as solid content, density grade, sand/binder ratio, and silica fume dosage on workability and strength properties of the CHPB matrix.

2. Experimental Program

2.1. Materials

2.1.1. Binder. Ordinary Portland cement (OPC) is the most common binder agent used in backfill operations due to its versatility. However, the optimization of filling material performance and the reduction of backfill cost lead to the extensive use of some pozzolanic materials, such as fly ash (FA) and blast furnace slag (slag) [19, 20].

OPC was used as binder and blended with silica fume at different replacement ratios in this study. The principal binder used is OPC (Type P.O 42.5 R, from Xuzhou, Jiangsu Province, China, based on Chinese standard GB175-2007 common Portland cement). Silica fume (from Chengdu City, Sichuan Province, China) is mainly composed of silica with a particle size of about 0.15 μm; hence, silica fume is often used as an active mineral admixture. The chemical composition of OPC and silica fume is presented in Table 1, which was measured by the X-ray fluorescence spectrometer (XRF).

2.1.2. Surface Sand. Sand used in the study was taken from the surface of Menkeqing coal mine, Ordos City, Shendong mining area, Inner Mongolia Region of China, where is rich in surface sand. Particle size distribution of the sample was determined using a laser particle size analyzer, as shown in Figure 1. The sampled surface sand has a relatively poor particle size distribution with a uniformity coefficient (Cu) of 2.5 and with particles over 0.1 mm accounted for 82.15% (over 75%), which belongs to fine sand. Meanwhile, the sample had an initial water content of around 5% and bulk density of 1.704 g/cm³. Moreover, the crystalline components in tailings were identified using X-ray diffraction (XRD). XRD patterns displayed in Figure 2 show that the crystalline phases are predominantly quartz, albite, and potassium feldspar, which are all inert substances.

2.1.3. Foaming Agent and Water. HTQ-1 compound foaming agent developed by Huatai Building Materials Company, Henan province, China, was used in this study, of which the main raw materials are nonpolluting animal protein oil and vegetable oil and will not produce any side effects to producers, users, and the environment. Meanwhile, the foaming agent can be used under the low temperature environment. The foaming agent adopts the physical method for foaming (Figure 3(a)), which has high foaming multiple and good stability. At the same time, the bubbles are independent of each other and evenly distributed in the cement slurry, which can form a large number of closed pores (Figure 3(b)).

Tap water was used to mix the binders and fine aggregate and used for foaming solution, with a pH of 6.8.

2.2. Sample Preparation and Mix Proportion. CHPB samples with different solid contents, surface sand/binder (S/B) ratios, density grade, and silica fume in binder were prepared. First, prepare the raw materials according to the experimental project. Second, the mixture slurry was prepared by using a blender, and after diluting the foaming agent 30 times, stable foam was formed through the foaming mechanism. Then, the foam was added to the mixture slurry and stirred well to produce the desired CHPB mixture (Figure 4). The produced backfill mixtures were cast into 100 × 100 × 100 mm cubic molds. Demoulded after 24 hours, the manufactured cubes were cured in the environmental chambers (Figure 5) at specific curing temperatures for periods of 7, 28, and 90 days. The relative humidity and the temperature were kept constant at %95 ± 1 and 20°C ± 2, respectively. The detailed testing protocols of the CHPB mixtures are summarized in Table 2.

It should be noted that the weight of samples of the same density grade should not exceed ± 50 g from the design value due to the different degree of damage of the foam in the stirring process.

2.3. Experimental Testing

2.3.1. Workability Tests. The workability of the fresh CHPB mix was evaluated by measuring the spread flow using a slump cone with the following dimensions: top diameter, 100 mm; bottom diameter, 200 mm; height, 300 mm. The
The estimated spread value was calculated as the average of the two measurements (Figure 6).

$$W = \frac{m_1}{(M - m_0)(1 - C_m)}$$  \hspace{1cm} (1)

where $C_m$ is the solid content.

2.3.2. Unconfined Compressive Strength Tests. At the predetermined curing periods (7, 28, and 90 days), unconfined compressive tests were performed on the cubic samples with a computer-controlled loading machine (Figure 7). The UCS tests were carried out at a constant deformation rate of 0.5 mm/min. Before conducting the UCS tests, the end surfaces of samples were polished to make sure that they are flat and parallel. All the measurements were carried out in triplicate, and the average values were presented in the results. The individual strengths of three specimens, molded with the same characteristics, should not deviate by more than 15% from the mean compressive strength; otherwise, an additional set of specimens needs to be prepared and tested again.

2.3.3. Microstructural Tests. Additional tests, including thermal analyses and specimen microstructure measurement, were carried out to better assess and interpret the results of the UCS tests. The phase compositions of the binder hydration products were determined by using thermal analyses (thermogravimetry (TG) and differential thermogravimetry (DTG)). TG/DTG analyses were performed on specimens at 7 d, 28 d, and 90 d for quantitative analysis of hydration products in different curing time with a PerkinElmer TGA-8000 (Figure 8) in a dynamic N2 atmosphere at a heating rate of 10°C/min. SEM studies were done with a Quanta TM 250 operated at an accelerating voltage of 15 kV on some CHPB samples. After being soaked in anhydrous ethanol to terminate the hydration process, the specimens were taken out, dried for gold spray treatment, and put into SEM for microstructure analysis.

3. Result and Discussions

3.1. Effect of Density Grade. The overall morphology and the microstructure of specific hydration products of the CHPB specimen at 83.7% solid content cured for 28 days were scanned by the scanning electron microscope. The internal structure of the specimen is similar to honeycomb, which is mainly composed of closed pores with uniform distribution and irregular shape (Figure 9(a)). The dense structure formed by the binder wrapping surface sand is the part outside the pores (Figure 9(b)). The bonding interface of surface sand is filled with hydration products, which mainly consisted of flocculent calcium silicate gel, a certain amount of fine needle-like ettringite, and hexagonal flake calcium hydroxide crystals (Figure 9(c)).
The Group D of fresh CHPB samples were prepared to investigate the influence of density grade on workability and strength development. The effect of density grade ranging from 600 kg/m$^3$ to 1600 kg/m$^3$ for CHPB mixtures (solid content of 83.7%, surface sand/binder of 3.5, and silica fume in binder of 10%) is illustrated in Figure 10. It is clear that density grade has a significant effect on the workability and strength of the sample when other ratios are constant.

As shown in Figure 10(a), when increasing density grade from 600 kg/m$^3$ to 1600 kg/m$^3$, the corresponding spread decreases by 11.5%, 9.2%, 8.2%, 5.5%, and 4.3% and the decreasing range of spread gradually slows down. At the same time, the bleeding rate of the fresh mix was 4.5%, 5.1%, 4.8%, 5.2%, 4.1%, and 3.9%, respectively. It indicates that the density grade has little effect on the bleeding rate and the mix slurry with solid content of 83.7% and surface sand/binder of 3.5, and silica fume in binder of 10% has good cohesion.

The sample strength curve with different curing times (i.e., 7 days, 28 days, and 90 days) are approximately an exponential function shown in Figure 10(b). The UCS of CHPB with 28 days curing time from 600 kg/m$^3$ to 1600 kg/m$^3$ were 0.62 MPa, 1.00 MPa, 1.57 MPa, 3.23 MPa, 8.63 MPa, and 12.00 MPa, respectively. The results shown in Figure 10(b) underline that the curing time has a significant positive role in UCS and its development of CHPB samples.

For example, the UCS of CHPB samples with density grade of 1200 kg/m$^3$ increased from 1.95 MPa and 3.25 MPa to 3.88 MPa with increasing curing time from 7, 28, to 90 days. This increase in hydration products with curing time in the result of the TG/DTG analysis performed on 7, 28, and 90 days CHPB samples is shown in Figure 11. The DTG curves show a mass loss at 90–120°C, 440–460°C, and 720–730°C, which denote the evaporation of water along with dehydration of calcium silicate hydrate (C-S-H) and ettringite, dehydroxylation of portlandite (calcium hydroxide (C-H)), and decomposition of calcium carbonate, respectively [25]. A comparison of the DTG curves of the 7, 28, and 90 days CHPB samples shows that the peak or weight loss at around 440–460°C is higher for the latter sample. This indicates that the amount of hydration products increases with curing time since the amount of portlandite is an indication of the degree of hydration.

3.2. Effect of Solid Content. The group A of CHPB samples was prepared with different solid contents (i.e., 82.1%, 83.7%, 85.3%, and 87.0%) at density grade of 1000 kg/m$^3$, surface sand/binder of 3.5, and silica fume in binder of 10%. The effect of solid content on workability and strength development of CHPB samples is illustrated by Figure 12. As
expected, the workability of fresh CHPB samples obviously decreases as the solid content rises (Figure 12(a)).

This decrease in workability with a higher solid content (lower water content) is due to the reduction in interparticle distance, thus making it easier for the solid particles to slide past one another during shearing [26, 27].

The results presented in Figure 12(b) indicate that all of the samples show a similar tendency in the time-dependent changes of the UCS. It can be observed that when the solid content is less than 83.7%, the strength gain of CHPB samples increases with increasing solid content. This can be attributed to the content of calcium silicate hydrate (C-S-H) gel and total porosity within CHPB increasing and decreasing, respectively, with higher solid content and then resulting in denser structure and higher strength [28]. When the solid content is higher than 83.7%, the strength of CHPB samples shows a trend of negative growth with solid content increasing. This can be due to the excessively high solid content which means that the water-cement ratio is too low, resulting in insufficient hydration reaction and reducing in hydration products.

In addition, Figure 12(b) also illustrates that the UCS of CHPB increases with the elapsing of curing age regardless of solid content.

### Table 2: Summary of the mixture composition of the samples.

| Group | Sample | Solid content (wt. (%)) | Solid content proportions | Density class (kg·m⁻³) | Silica fume in binder (wt. (%)) |
|-------|--------|------------------------|--------------------------|------------------------|--------------------------------|
| A     | 1      | 82.1                   | 22.2                     | 77.8                   | 1000                           | 10                             |
|       | 2      | 83.7                   | 22.2                     | 77.8                   | 1000                           | 10                             |
|       | 3      | 85.3                   | 22.2                     | 77.8                   | 1000                           | 10                             |
|       | 4      | 87.0                   | 22.2                     | 77.8                   | 1000                           | 10                             |
| B     | 5      | 83.7                   | 25                       | 75                     | 1000                           | 10                             |
|       | 6      | 83.7                   | 22.2                     | 77.8                   | 1000                           | 10                             |
|       | 7      | 83.7                   | 20                       | 80                     | 1000                           | 10                             |
|       | 8      | 83.7                   | 18.2                     | 81.8                   | 1000                           | 10                             |
| C     | 9      | 83.7                   | 22.2                     | 77.8                   | 1000                           | 0                             |
|       | 10     | 83.7                   | 22.2                     | 77.8                   | 1000                           | 5                             |
|       | 11     | 83.7                   | 22.2                     | 77.8                   | 1000                           | 10                             |
|       | 12     | 83.7                   | 22.2                     | 77.8                   | 1000                           | 15                             |
| D     | 13     | 83.7                   | 22.2                     | 77.8                   | 600                            | 10                             |
|       | 14     | 83.7                   | 22.2                     | 77.8                   | 800                            | 10                             |
|       | 15     | 83.7                   | 22.2                     | 77.8                   | 1000                           | 10                             |
|       | 16     | 83.7                   | 22.2                     | 77.8                   | 1200                           | 10                             |
|       | 17     | 83.7                   | 22.2                     | 77.8                   | 1400                           | 10                             |
|       | 18     | 83.7                   | 22.2                     | 77.8                   | 1600                           | 10                             |

This decrease in workability with a higher solid content (lower water content) is due to the reduction in interparticle distance, thus making it easier for the solid particles to slide past one another during shearing [26, 27].

The results presented in Figure 12(b) indicate that all of the samples show a similar tendency in the time-dependent changes of the UCS. It can be noticed that when the solid content is less than 83.7%, the strength gain of CHPB samples increases with increasing solid content. This can be attributed to the content of calcium silicate hydrate (C-S-H) gel and total porosity within CHPB increasing and decreasing, respectively, with higher solid content and then resulting in denser structure and higher strength [28]. When the solid content is higher than 83.7%, the strength of CHPB samples shows a trend of negative growth with solid content increasing. This can be due to the excessively high solid content which means that the water-cement ratio is too low, resulting in insufficient hydration reaction and reducing in hydration products.

In addition, Figure 12(b) also illustrates that the UCS of CHPB increases with the elapsing of curing age regardless of solid content.

### 3.3 Effect of Sand/Binder Ratios.

The effects of sand/binder ratios on the workability and strength of the CHPB specimens as a function of curing time are summarized in Figure 13. In this test, four different sand/binder ratios (3.0, 3.5, 4.0, and 4.5) were prepared, while other parameters such as the density grade, solid content, and silica fume in binder were maintained constant at 1000 kg/m³, 83.7%, and 10%, respectively. It can be observed from Figure 13(a) that the spread of fresh CHPB increases slightly with an increase in the S/B ratio. That is because with the increase of the S/B ratio, the amount of sand gradually increases and the content of cementing material decreases. Due to the constant water
Figure 9: SEM diagram of the specimen at 83.7% solid content 28 days. (a) ×36; (b) ×500; (c) ×3000.

Figure 10: Effect of density grade on the (a) flow and strength of CHPB samples.

Figure 11: TG/DTG diagrams for the CHPB specimen at the density grade of 1200 kg/m$^3$ cured for 7, 28, and 90 days.
content, on the one hand, the corresponding water-binder ratio increases, and on the other hand, the encapsulation of cementing materials to sand decreases.

From Figure 13(b), it is obvious that the variation of the S/B ratio from 3 to 4.5 results in a decrease of the strength of CHPB. It is mainly due to the reduction of the relative content of cementing materials, resulting in the reduction of hydration products and the encapsulation force of mortar. At the same time, the increase of the water-binder ratio also reduces the compactness of structure. The results presented in Figure 13(b) indicate that all of the samples have a rapid increase in the UCS until 28 days and then a steady increase. This is due to the fact that cementitious materials develop strength with continual hydration. The rate of strength gain is faster at the early ages but reduced with curing time [29].

3.4. Effect of Silica Fume in Binder. Figure 14 demonstrates the strength development of CHPB samples produced by using a silica fume in binder of 0–15 wt%. The laboratory tests were conducted at a fixed solid content of 83.7%, a sand/binder ratio of 3.5, and a density grade of 1000 kg/m³. From Figure 14(a), it is fairly obvious that the spread of fresh CHPB is slightly improved with an increase in silica fume.
dosage in binder. An explanation for this behavior is that higher replacement of processing silica fume results in a reduction of friction between particles [30].

The results presented in Figure 14(b) underline that silica fume dosage in binder has a significant impact on the strength of CHPB. The UCS increases with the dosage up to 10% and then starts to decrease with higher dosage. The optimum dosage that produces the highest compressive strength of the CHPB appears to be about 10%. This can be explained by the following: (i) silica can react with calcium hydroxide (C-H) to form calcium silicate gel. At low dosage, an increase in silica fume leads to more amount of hydration products. (ii) The particle size of silica fume is smaller than that of cement, which can increase the compactness of the internal structure.

Subsequently, this strength is observed to deteriorate gradually as the dosage increase from 10% to 15%.

The reasons for the contrasting trend could be due to excessive silica powder which will reduce the amount of water involved in the hydration reaction, resulting in a decrease in the amount of early hydration products, unable to generate sufficient amount of C-H. At the same time, the agglomeration phenomenon occurs, which reduces the compactness of the material and causes its strength to decrease.

4. Conclusions

In this paper, a comprehensive laboratory test work has been conducted to study the influences of four parameters (i.e., solid content, sand/binder ratios, density grade, and silica fume dosage in binder) on workability and strength development of CHPB samples. For given CHPB samples with different density grades (i.e., 600 kg/m³, 800 kg/m³, 1000 kg/m³, 1200 kg/m³, 1400 kg/m³, and 1600 kg/m³), sand/binder ratio (i.e., 3, 3.5, 4, and 4.5), solid content (i.e., 82.1%, 83.7%, 85.3%, and 87.0%), and silica fume dosage in binder (i.e., 0%, 5%, 10%, and 15%) were prepared to obtain the workability and UCS at different curing times (i.e., 7 d, 28 d, and 90 d). Based on the results of the experimental tests performed, the following conclusions can be drawn:

(1) The internal structure of the CHPB sample contains a large number of closed pores. With the increase of the density grade, the number of pores decreases, and the UCS of the sample increases exponentially. On the contrary, the workability of the samples decreases significantly.

(2) Workability of fresh CHPB samples significantly decreases with increasing solid content due to the reduction of interparticle distance. The UCS increases with increasing solid content from 82.1% to 83.7% through a denser structure. However, a decrease trend in UCS is observed as the solid content exceeds 83.7%.

(3) The variation of the sand/binder ratio from 3 to 4.5 results in a slight increase of workability and a significant increase of the strength. It is mainly due to the reduction of the relative content of cementing materials, resulting in the reduction of hydration products. At the same time, the increase of the water-binder ratio also reduces the compactness of the structure.

(4) Adding silica fume poses a positive effect of silica fume dosage on workability behavior and UCS of CHPB samples. The optimum dosage in binder that produces the highest compressive strength of the CHPB appears to be about 10%.

Data Availability

The data used to support the findings of this study are available from the corresponding author upon request.
Conflicts of Interest
The authors declare that they have no conflicts of interest.

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