Vertical and bevel-structured SiC etching techniques incorporating different gas mixture plasmas for various microelectronic applications

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This study presents a detailed fabrication method, together with validation, discussion, and analysis, for state-of-the-art silicon carbide (SiC) etching of vertical and bevelled structures by using inductively coupled plasma reactive ion etching (ICP-RIE) for microelectronic applications. Applying different gas mixtures, a maximum bevel angle of 87° (almost vertical), large-angle bevels ranging from 40° to 80°, and small-angle bevels ranging from 7° to 17° were achieved separately using distinct gas mixtures at different ratios. We found that SF$_6$ with additive O$_2$ was effective for vertical etching, with a best etching rate of 3050 Å/min. As for the large-angle bevel structures, BCl$_3$ + N$_2$ gas mixtures show better characteristics, exhibiting a controllable and large etching angle range from 40° to 80° through the adjustment of the mixture ratio. Additionally, a Cl$_2$ + O$_2$ mixture at different ratios is applied to achieve a small-angle bevels ranging from 7° to 17°. A minimum bevel angel of approximately 7° was achieved under the specific volume of 2.4 sccm Cl$_2$ and 3.6 sccm O$_2$. These results can be used to improve performance in various microelectronic applications including MMIC via holes, PIN diodes, Schottky diodes, JFETs’ bevel mesa, and avalanche photodiode fabrication.

Silicon carbide (SiC) is a wide bandgap compound semiconductor with excellent thermal conductivity, high electric breakdown voltage, and high-temperature stability, making it a good material for high-power, high-frequency, or high-temperature electronic devices, such as Schottky diodes, field effect transistors (FETs), and high-efficiency light-emitting diodes. However, SiC’s strong internal bonding energy results in a high chemical resistance, a property which eventually restricts the etching rate of the mask, and its hardness (H = 9 on the Mohs scale) makes chemical etching in traditional solutions difficult. Because of its properties, SiC can also be used extensively for growth of an epitaxial layer of gallium nitride (GaN), used to form high-quality substrates for high-power monolithic microwave integrated circuit (MMIC) devices such as power amplifiers, low-noise amplifiers, and mixers. This is especially convenient as SiC is susceptible to dry etching, producing SiC-based MMIC devices through backside vertical via-hole etching.

The proposed fabrication technology provides a basis for the future development of a wide variety of SiC-based devices. Various dry etching techniques have been used both to address the etching rate issue and ensure excellent selectivity. Among the various dry etching techniques hitherto applied, inductively coupled plasma reactive ion etching (ICP-RIE) is the most widely adopted technique featuring a damage-free, highly anisotropic and selective, with a high etching rate. It also allows independent adjustment of the gas mixture and the flow rate.
A variety of fluorine-, chlorine-, and bromine-based plasma chemistries—including NF₃, NF₃/O₂, SF₆/O₂, SF₆/He, SF₆/O₂/Ar, ICl, IBr, Cl₂/Ar, and BCl₃/Ar—have been studied for SiC etching. Among them, fluorine-based chemistries are the most effective gases when compared to other mixtures in terms of their ease of implementation and etching rate. In general, SiC dry etching is conducted using fluorine radicals as a primary source, followed by additive gases such as O₂, Ar, N₂, and H₂ as a secondary source—to control and enhance the etching process. The highest etching rate can be achieved with both SF₆ and NF₃, because of their rapid dissociation in plasma; however, SF₆ is the preferred feed gas based on cost and safety considerations.

Recently, researchers have put plenty of efforts into developing various microelectronic devices based on the bevel structure. Large-angle bevel structure is commonly used in the application of PIN diodes, Schottky diodes, static induction transistors (SITs), and junction FETs (JFETs). To maintain the high performance and reliability of these devices, a smooth and accurate bevel angle should be guaranteed so that the issue of electric field concentration can be efficiently solved. Several SiC diodes, transistors, and switches with high breakdown voltages (greater than 10 kV) have been reported, using a mesa shape with junction termination extensions formed by a notably large-angle bevel structure for SiC etching, which contributes to improve the breakdown voltage by alleviating the electric field crowding at the device edges. For small-angle bevel, it plays an important role in the application of avalanche photodiodes (APD). Particular concern has been focused on the mesa structure with bevel sidewalls instead of vertical sidewalls. It can be referred that bevelling the sidewalls suppresses edge breakdown. The effect of bevelled sidewalls is to increase the depletion width at the surface of the device, and therefore improve the breakdown voltage of the APD.

The use of conventional SF₆ + O₂ gas mixture for bevel etch of the SiC material is limited by the difficulties in controlling both the mixture ratio and the RF power. Accordingly, chlorine-based gas mixtures are carried out and have become notably promising for obtaining a smooth surface on the epitaxial layer. Most importantly, an independently controls of system parameters can be realized.

In this work, the vertical etching of SiC using SF₆/O₂ plasma material with different ratios have been studied, resulting in an improved etch rate of 3050 Å/min. We have also found Ni to be a robust etch mask material, allowing a high selectivity of 100:1. Bevel structures with a large angle ranging from 40° to 80° were studied particularly under different gas mixtures, including BCl₃/Cl₂ and BCl₃/N₂, in which BCl₃/N₂ shows the best results in terms of obtaining controllable angles through adjusting the ratio of BCl₃ with (BCl₃ + N₂). Large-angle bevels in SiC etching are significant for the growth of the epitaxial layer of AlGaN/GaN/SiC diodes, transistors, and switches. According to a preferred embodiment of the concept, an angle of approximately 60° optimizes the performance of the epitaxial layer. For other applications, smaller angles of approximately 40° are preferable, as they are less prone to peripheral breakdown.

Small-angle bevel often encounters problems with early edge breakdown, as a locally enhanced electric field occurs on the etched junction surface at the active region of APDs during the mesa etching. The formation of a small (~7°) SiC sidewall bevel structure is favoured to prevent such a premature edge breakdown. Small-angle bevel structures (from 7° to 15°, in particular) are achieved using Cl₂/O₂ gas chemistry, which has been found in miniature, highly reliable, low power consumption, SiC-based APDs.

**Methods**

This study combined various experiments to optimize the etching rate and selectivity on the 4H-SiC substrate. The performance of two different metal masks (Ni and Cr) was compared under a fixed SiC etching rate of 3050 Å/min. The more robust Ni mask, which allows for improved selectivity, was used as an etching mask in our other experiment. These masks were built to a thickness of 11 µm through e-beam evaporation at a deposition rate of 5 Å/sec and under a vacuum of 3.5 E⁻⁶ Torr.

Another experiment investigated the effects of different gas mixtures on the formation of various angles and bevel structures, ranging from as low as 7° (which can be effectively applied to APDs), 40°–80° (applicable for PIN diodes, Schottky diodes, and JFETs’ bevel mesa), and up to as high as 90° (interpreted as vertical etching, for application in MMICs through backside via-hole etching).

Further experimentation evaluated the impact of mixture ratios (from 0 to 100%) in SF₆/O₂ gas on vertical bevel structure, using a fixed coil power of 2000 W and platen power of 200 W. The applied pressure was set up constantly at 20 mTorr, and the etching time was 5 h for all samples. Six samples were tested, with SF₆ gas flow rates varying from 0 sccm to 25 sccm at 5 sccm intervals, while the O₂ flow rate was inversely adjusted.

This study also developed an effective formation process for large-angle bevel-structured etching using BCl₃ + N₂ and BCl₃ + Cl₂ gas mixtures. Different conditions for SiC etching were studied using BCl₃, with ratios ranging from 0 to 100%. Fixed process conditions (coil power = 900 W, platen power = 300 W, process pressure = 5 mTorr, and etching time = 30 min) were used for the ICP dry etching process with both BCl₃ + N₂ and BCl₃ + Cl₂. Furthermore, different mixtures of gases—including BCl₃, BCl₃ + Cl₂, Cl₂, and Cl₂ + O₂ at different ratios—were studied as a means to obtain small-angle bevel structures. In total, six different tests were conducted, including BCl₃ alone at a volume of 6 sccm, BCl₃ + Cl₂ at volumes of 4.8 sccm BCl₃ + 1.2 sccm Cl₂, Cl₂ alone at a volume of 6 sccm, and Cl₂, O₂ at volumes of 4.8 sccm Cl₂, 1.2 sccm O₂, 3.6 sccm Cl₂, 2.4 sccm O₂, and 2.4 sccm Cl₂, 3.6 sccm O₂. A detailed summary of this experiment is illustrated in Table 1.

**Results and Discussion**

**Vertical Etching.** Figure 1(a) demonstrates the changes in SiC etching rate when performed with different ratios of SF₆ + O₂. The highest SiC removal rate, 3050 Å/min, is achieved at the SF₆ mixing ratio of 80%, whereas the etching rate tends to decrease for mixture ratios beyond 80%. In the reactive F⁻ ions, reactive gas dilution, removal efficiency of the etching products, decreased sulphur reaction efficiency, and competition from forming SiO₂, likely all combine to result in the increase and eventual decline of the etching rate. Although the addition of O₂ to the SF₆ plasma provides another pathway for volatilizing C in the forms CO, CO₂, etc.—thereby increasing
the SiC etching rate—it also produces SiO\textsubscript{2} on the surface, which can limit the etching process. As a result of this competition, an optimum O\textsubscript{2} ratio in the SF\textsubscript{6}/O\textsubscript{2} gas mixture of around 20% is obtained.

The cross-sectional images of the etched SiC, based on the SF\textsubscript{6} gas ratio of 40%, 60%, and 80%, are illustrated in Fig. 1(b) to show well-defined vertical etch structures, which verify the success of our proposed experiments and demonstrate the observed differences dependent on our proposed work from all of the etching conditions.

Figure 1(c) summarizes detailed measurement results of the Ni and Cr applied in this work. The etching rates observed when using Ni and Cr masks are 30 Å/min and 77 Å/min, respectively, when all other parameters are held constant to allow a SiC etching rate of 3050 Å/min. This discrepancy results in a selectivity of 100:1 for SiC:Ni and 40:1 for SiC:Cr. Because of this, the SF\textsubscript{6}-optimized etching conditions for SiC exhibit higher etching rates when Ni metal masks are used, approximately 3050 Å/min with a selectivity of 100:1. This was experimentally verified in Fig. 1(d).

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**Table 1.** Summarization of the detailed information of all experiments conducted in this work. All of the experiment is based on 4H-SiC substrate, Ni mask is applied in the process of vertical etching and large-angle bevel etching, and AZ4620 photoresist is used as the etching mask for small-angle bevel etching.

| Gas Mixture | Vertical Etching | Large-angle Bevel Etching | Small-angle Bevel Etching |
|-------------|-----------------|---------------------------|---------------------------|
| SF\textsubscript{6}, O\textsubscript{2} | BCl\textsubscript{3}, N\textsubscript{2} | BCl\textsubscript{3}, Cl\textsubscript{2} | BCl\textsubscript{3}, Cl\textsubscript{2}, Cl\textsubscript{2}, O\textsubscript{2} |
| Totally 25 sccm | Totally 40 sccm | Totally 6 sccm |

| System Condition | Coil Power | Platen Power | Applied Pressure | Etching Time |
|------------------|------------|-------------|-----------------|-------------|
| Coils Power: 2000 W | 200 W | 300 W | 5 mTorr | 5 h |
| Coils Power: 900 W | 300 W | 3 mTorr | 30 min | 30 min |
| Coils Power: 0 W | 200 W | 200 W | 3 mTorr | 30 min |

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**Figure 1.** Etching and selectivity profile of the SF\textsubscript{6} + O\textsubscript{2} gas mixture. (a) Etching rate characteristics of the SF\textsubscript{6} + O\textsubscript{2} gas for different ratios. (b) Cross-sectional image of the etched SiC based on different gas ratio. (c) Etching rate and selectivity results for Ni and Cr metal masks, with a fixed SiC etching rate of 3050 Å/min. (d) AFM image of the etching surface at different ratios of SF\textsubscript{6} + O\textsubscript{2} gas.
verified using a gas mixture of SF\textsubscript{6} (20 sccm) + O\textsubscript{2} (5 sccm). This preferential selectivity occurs because both the Ni and Cr metals react with fluoride gas, respectively forming NiF\textsubscript{2} and CrF\textsubscript{3}, and the by-product NiF\textsubscript{2} is more stable than CrF\textsubscript{3} in air. As NiF\textsubscript{2} is less volatile (the sublimation temperature of NiF\textsubscript{2} is 1474 °C, whereas the anhydrous form of CrF\textsubscript{3} sublimates at 1100–1200 °C), it causes less damage to the surface during etching, resulting in the higher selectivity\textsuperscript{29, 30}. The slow etching rate observed with the Ni mask demonstrates that Ni is versatile enough be used as mask under various etching conditions. In contrast, Cr masks work only in fluorine-free environments, which minimize mask erosion.

Atomic force microscopy (AFM) images are exhibited in Fig. 1(d), illustrating surface roughness after the following etching of gas mixtures with different ratios. It can be observed that surface roughness becomes less smooth with increasing of the SF\textsubscript{6} proportion in SF\textsubscript{6} + O\textsubscript{2}. As the ratio of SF\textsubscript{6} increases, the surface roughness gradually deteriorates from 6.7 nm to 11.3 nm–14.0 nm before suddenly rebounding to 22.1 nm when the flow rate of SF\textsubscript{6} is 25 sccm. The deterioration in the surface roughness occurs primarily because low-volatility reaction products, such as CF\textsubscript{x} (CF\textsubscript{2}, CF\textsubscript{3}), are generated and scattered across the etching surface as a secondary mask. Generally, when more fluorine ions participate in the reaction, more products are created and etching surface becomes rougher. In addition, serious roughness deterioration occurs when the gas ratio is over 80%, this is due to the fluorine atoms are of primary interest and they act as the main chemical reagents. The fraction of free surface increases rapidly, providing favourable conditions for both the physical sputtering etching and the chemical reaction. The optimized parameters used to achieve the 3050 Å/min etching rate with ICP dry etching are: coil power = 2000 W, platen power = 200 W, processing mixture = SF\textsubscript{6} (20 sccm) + O\textsubscript{2} (5 sccm), and processing pressure = 20 mTorr. Figure 2 shows a series of scanning electron microscopy (SEM) images of a 70-µm via-hole array formed on a SiC substrate, as obtained in this study.

Large-angle Bevel Etching. Even though the SF\textsubscript{6}/O\textsubscript{2} gas mixture can achieve a relative high etching rate, it is not the ideal solution for bevel etching; the SiC etching profile remains unchanged through variations in component ratio of the SF\textsubscript{6}/O\textsubscript{2} gas mixture. This is because the most impactful species in SF\textsubscript{6}/O\textsubscript{2} plasma processing is created when electrons collide with neutral gas molecules. These collisions result in dissociation (leading to radical formation), ionization (SF\textsubscript{6}\textsuperscript{+}), and excitation, in accordance with the energy required for each process\textsuperscript{31}. In an SF\textsubscript{6}/O\textsubscript{2}-SiC etching system, the O\textsubscript{2} passivates the SiC surface with a SiO\textsubscript{2} layer; subsequently, SF\textsubscript{6}\textsuperscript{+} ions etch these passivations and allow the F\textsuperscript{+} radicals to etch the SiC substrate beneath. Thus, anisotropic SiC etching is achieved.

Given this process, a BCl\textsubscript{3}-based etching atmosphere is considered for bevel SiC etching. Figure 3(a) demonstrates the SiC etching rate changes according to the contents of the processing gas; as the total flow rate of the processing gas is kept constant at 40 sccm, the content ratio of the N\textsubscript{2} or Cl\textsubscript{2} gas is adjusted to obtain various ratios (from 0 to 100%) of the whole gas mixture. During the BCl\textsubscript{3} + Cl\textsubscript{2} gas mixing process, the etch rate is observed to be inversely proportional to the content of BCl\textsubscript{3}. This occurs because the addition of chloride gas reduces the densities of positive ions and electrons. As a consequence of this higher dissociation threshold energy, BCl\textsubscript{3} can absorb more energy before molecular dissociation than Cl\textsubscript{2}\textsuperscript{32}. The highest SiC removal rate (1330 Å/min) is observed when the BCl\textsubscript{3} + Cl\textsubscript{2} gas mixture is held with a BCl\textsubscript{3} content ratio of 0%. The removal rate decreases dramatically to 407 Å/min at a BCl\textsubscript{3} content ratio of 100%.

Figure 2. SEM images of various via-holes formed on a SiC substrate using optimized parameters for the ICP-RIE technique. (a) Mask opening widths of 35 μm, 25 μm, 20 μm, 100 μm, 70 μm, 70 μm (from the top left, row first). (b) Vertically etched SiC with a complicated pattern and a well-etched sidewall profile. (c) Mask opening width of 70 μm (circular) and 70 μm (square), from left to right. (d) Circular pattern enlarged five times. (e) Circular pattern enlarged ten times.
In contrast, the SiC etching rate when using a BCl₃ + N₂ gas mixture does not vary consistently with BCl₃ ratio. Instead, the etching rate increases almost linearly as the BCl₃ content ratio increases from 0 to 60%, but decreases significantly and abruptly beyond this point. The highest removal rate (860 Å/min) was observed at a 60% content ratio, a rate which falls to 407 Å/min as the BCl₃ content ratio reaches 100%. The addition of BCl₃ to N₂ plasma initially results in a significant increase in etching rate, caused by increased BCl₃ dissociation. The dissociation of BCl₃ stems from an increase in electron temperature, the result of energy transferred from the N₂ metastable. As such, the N₂ metastable is responsible for the both the increased dissociation and enhanced production of the etch species. However, the continuous increase in BCl₃ content ends up reducing the etching rate, which is most likely the result of a decrease in the effective volume density of the BCl₃-ions and reactive Cl⁻. Such a decrease reduces the ion bombardment and associated chemical reaction between the SiC substrate and Cl⁻-atoms while the ICP power is maintained; thereby reducing the number of reactive species and ultimately leading to a decrease in etching rate.

In accordance with previously reported quadrupole mass spectrometry results, BCl₂, BCl₃, Cl₂, and Cl⁻ will exist as a function of the percentage of N₂ in the flow. The Cl₂ intensity can increase up to several times the initial value for BCl₃ in the presence of 60% N₂. Under these conditions, the Cl⁻ intensity will decrease, but continues to exhibit a local maximum at 80%. The recombination tendency of Cl⁻, reforming Cl₂ during transport to the mass spectrometer, may account for the low measurements Cl⁻ intensity that result in the observed high SiC etching rate.

Figure 3(b) shows the obtained etching profile angle as a function of the gas mixture ratio. Large etching angles—ranging from 70° to 77°—are achieved by applying the processing gas mixtures BCl₃ + Cl₂, whereas in the case of BCl₃ + N₂ the angle tends to increase with the gas mixture ratio. An etching profile angle of approximately 40° is noted at a BCl₃ + N₂ gas mixture ratio of 20%, and it increases to approximately 75° at a BCl₃ gas mixture ratio of 100%. However, the use of a BCl₃ + Cl₂ gas mixture plasma results in (almost perfect) smooth vertical walls with relatively good anisotropy, because of the continual existence of the diboron tetrachloride (B₂Cl₄) thin polymer layer, which produces sidewall passivation. We believe that the narrow-range profile angle changes are caused by the generation of chloropolymers on the freshly etched SiC surface, resulting in B₂Cl₄ that cannot be easily etched with BCl₃ and/or Cl₂. The controllable angle characteristics of BCl₃ + N₂ gas mixtures can be attributed to the results of anisotropic etching and N₂-promoted passivation. First of all, the mask edges might be oxidized by the pristinely etched SiC surface generated by the O₂ residual, resulting in SiOₓ thin films. The oxygen sources can be residual oxygen gases or reaction by-products from the hard mask. As soon as N₂ (<40%) is added to the BCl₃ mixture gas, a slight compound-nitride-like passivation layer is deposited. At this point, the SiC sidewall will remain close to vertical, showing the SiOₓ layer is thick and dense enough to prevent chemical etching reactions between neutral chlorine species and the sidewall. With the continuous increase of N₂ (>40%), passivation deposition becomes excessive, the etch rate decreases, and the sidewall becomes more extensively profiled. Moreover, a grass-like roughness appears at the bottom of the etched area, caused by the formation of a compound-nitride-like passivation layer, which requires a 1:6 buffered oxide etching process for thorough removal from the thin SiOₓ film. Similar SiOₓ and nitride-based passivation layers are observed on InP wafers, etched by BCl₃/N₂ and Cl₂/N₂ gas mixtures.

To investigate the effects of N₂ and Cl₂ in a BCl₃ plasma, gas flow ratio of the two gases are separately varied from 80% to 0%. Under this procedure, an evaluation of improvements in sidewall etching profile control can be conducted, observing the separate effects of N₂ and Cl₂ gas. Corresponding SEM images of etching profile angles are shown in Fig. 4(a1–a4) and Fig. 4(b1–b4). A wider angle range, from 41° to 78°, is observed when using N₂ + BCl₃ as the etching gas mixture; in comparison, using Cl₂ + BCl₃ results in larger angles within a more narrow range, from 70° to 77°. These findings show the BCl₃ + N₂ mixture is an optimal choice for wide-range, tuneable, large-angle SiC bevel etching.
Small-angle Bevel Etching. Small-angle bevelled mesa etching demands that focus be placed simultaneously on achieving small angles (~7°) and ensuring surface smoothness, a requirement for effective application in APDs. This prevents the premature breakdown around the mesa edge termination referred to at the beginning of the paper. Aiming for the smallest possible bevelled sidewall increases the depletion width at the surface of the APD, in comparison to bulk. As such, the electric field is lower at the edges than the centre, further contributing to quite high breakdown characteristics and allowing bulk breakdown to precede surface breakdown. In an SF6 + O2 mixture, the chance of polymer formation during photoresist etching increases greatly with the increase in O2 ratio. These polymers, once formed by plasma, function as very tough microetching masks that transfer to the SiC surface and eventually obstruct SiC etching. Additionally, by using SF6 more fluoride ions participate in the reaction, creating more by-products. These by-products are then responsible for generating roughness in the polymer surface, because of scattering on the etching surface as a secondary mask. This effect restricts the use of SF6 + O2 as a gaseous mixture for small-angle bevel etching. BCl3 could be another choice for achieving small-angle bevels; however, the anisotropy characteristics of BCl3 only allow large-angle bevels (≥70°), unlike some other mixture gases. As mentioned before, if a gas such as N2 is added to the BCl3 atmosphere, a relatively small-angle bevel (almost as low as 40°) may be formed. However, for the preparation of ultra-small SiC sidewall bevel angles, a larger etching selectivity between the etching mask and the etched object must be obtained, properly adjusted, and controlled.

In this study, a well-patterned photoresist is used as the etching mask due to its chemical activity, as shown in Figure S1 and S2 (supplementary section). Given that Cl2 is more inert to photoresist than BCl3, it enable etching of smaller angles (<20°); and with the addition of a regular photoresist reactant gas (i.e. O2), the mixture gas is likely to allow even smaller bevel angles. The smallest bevel etching angle (as low as 7.63°), is achieved by using a Cl2/O2 gas mixture with a gas flow ratio of 2.4 sccm for Cl2 and 3.6 sccm for O2 (shown in Fig. 5). This indicates that when the majority of the gaseous mixture is O2, photoresist erosion is significantly accelerated without inducing further SiC etching. The etching depth, etching rate, and selectivity thus obtained are 1.23 µm, 411 Å/m, and from 761 to 411 Å/min, respectively. The use of Cl2 gas alone at 6 sccm results in a maximum etching depth of 2.28 µm and an etching rate of 711 Å/min, whereas the minimum etching depth of 0.33 µm and an etching rate of 109 Å/min are obtained by using BCl3 gas at 6 sccm. The use of Cl2 as an etchant results in a small angle, whether alone or mixed with O2. As the ratio of O2 is increased with respect to Cl2 (from 0 to 3.6 sccm of O2), the bevel angle is found to decrease from 17.91° to 7.63°, at the expense of decreasing both the etching depth and etching rate, which decrease from 2.28 to 1.23 µm and from 761 to 411 Å/min, respectively. With the continuous increase of O2 from 4.8 sccm to 6.0 sccm, the etch selectivity of all samples increases monotonically with the increasing O2 flow rate, which shows that the patterned photoresist is not applicable as a suitable etching mask for SiC ICP-RIE. The surface roughness of the SiC substrate depends on the etching conditions, and therefore remains another important parameter when evaluating the small-angle bevel etching quality, particularly during the APD mesa termination fabrication. The SiC layer surface morphologies produced by different Cl2-related gas mixtures are shown in Fig. 6—via AFM—reveling that the use of Cl2 + O2 gas mixtures causes a slight change of surface morphology. In the absence of SiC etching, surface roughness is found to be 36.0 nm, and increases slightly in the presence of the various Cl2 + O2 gas mixtures. Using Cl2 alone, the surface roughness is found to be 62.8 nm. As the amount of O2 increases, the SiC surface roughness deteriorates from 67.4 nm to 78.4 nm. The surface roughness worsens with the formation of oxide films in the presence of excessive O2, greatly reducing the etching rate in certain areas.

Figure 4. SEM images of the etching profile angles obtained with different mixture ratios of BCl3 with (BCl3 + N2) labelled as (a) and (BCl3 + Cl2) labelled as (b), showing that etching angles can be controlled by adjusting the gas mixture ratio. (a1) Mixture ratio of 20% results in the lowest etching angle, 41.11°; (a2) mixture ratio of 40% results in an etching angle of 45.73°; (a3) mixture ratio of 80% results in an etching angle of 61.67°; and (a4) BCl3 gas alone (100%) results in the largest etching angle of 78.04°. (b1) Mixture ratio of 20% results in an etching angle of 76.45°; (b2) mixture ratio of 40% results in an etching angle of 72.62°; (b3) mixture ratio of 80% results in the lowest etching angle of 70.78°; and (b4) BCl3 gas alone (100%) results in the largest etching angle of 77.46°. 

Figure 6—via AFM—revealing that the use of Cl2 + O2 gas mixtures causes a slight change of surface morphology. In
However, when the O₂ content increases to 3.6 sccm, the surface roughness improves to 71.3 nm. Unevenly etched surfaces occur primarily because of either splits in the metal mask material and plasma polymer residues, which scatter throughout the etching region and form a microscopic mask. Oxygen plasma can help removing this scattered layer. Therefore, adding an appropriate amount of O₂ can improve the surface roughness.

Conclusions
In this work, different gas chemistries were used to obtain a variety of bevel angles for SiC substrates, and their performance for ICP-RIE etching was investigated and analysed. Results showed that a high etching rate and improved selectivity for vertical etching can be obtained with the use of a fluorine-based gas mixture (i.e. SF₆) in conjunction with different ratios of O₂ and a Ni mask. Additionally, it was found that a wide, controllable
range of bevel angles can be achieved in large-angle bevel structure formation by using a BCl₃ + N₂ gas mixture. It was also demonstrated that Cl₂ gas can effectively achieve small-angle bevel structures, and that the addition of different ratios of O₂ can further reduce the small-angle bevel angle to as low as 7°. Furthermore, it was shown that surface morphology is not significantly affected by the use of Cl₂ and O₂ etching gas mixtures, with a smooth surface being maintained on the SiC layer.

The vertical SiC etching process developed here can be applied to large volume manufacturing for future SiC backside etching and source grounding applications. Large-angle bevels in SiC etching are significant for the bevel mesa of diodes, transistors, and switches. Moreover, the issue caused by electric field concentration could be overcome through large-angle bevel structure, so that high performance and high reliability can be obtained for the devices including PIN diodes, Schottky diodes, STJs, and JFETs. Finally, with optimal small-angle SiC bevel structures and a smooth surface morphology, the leakage current of the APDs could be effectively reduced, and greatly improved breakdown properties will be achieved as predicted in other works.

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**Author Contributions**

Ho-Kun Sung performed the design and analysis of the device and wrote the manuscript; Hee-Kwan Lee, Bum-Doo Park, and Woong-Sun Lim contributed to the fabrication; Yang Li, Qun Wu and Kyung-Ho Park supervised the analysis and co-wrote the manuscript; Tian Qiang advised in preparation of the manuscript. As the corresponding authors, Zhao Yao and Cong Wang provided the overall research conception, guided the research, and revised the manuscript. All authors discussed the results and implications and commented on the manuscript at all stages. All authors have given approval to the final version of the manuscript.

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