Influence of gamma-ray irradiation on optical and thermal degradation of poly(ethyl-methacrylate) (PEMA) polymer

S. Fares

National Center for Radiation Research and Technology (NCRRT), Nasr City, Egypt; sfares2@yahoo.com

Received 23 April 2012; revised 26 May 2012; accepted 11 June 2012

ABSTRACT

Polymer based on poly(ethyl-methacrylate) (PEMA) were exposed to different doses of gamma radiation up to 100 kGy and the physical properties have been studied. The effects of gamma irradiation on the optical spectrum of PEMA films have been investigated using spectrophotometric measurements of reflectance and transmittance in the wavelength range 200 - 1100 nm. The structure of the sample is analyzed by X-ray diffraction technique and is found to be amorphous and partially crystalline. TGA studies revealed that the thermal stability of polyethyl methacrylate, improved after irradiation doses up to 100 kGy. On other hand driving absorption coefficient $\alpha(\omega)$, consequently the band tail width $E_t$ and optical band gap estimated. This behavior is believed to be associated with the generation of excess of electronic localized states. Also, Optical constants such as refractive index ($n$), extinction coefficient ($K$) have been determined using Swanepole method. Optical dispersion parameters and the dispersion parameters, such as $E_o$ (single-oscillator energy), $E_d$ (dispersive energy) are discussed in terms of the single-oscillator Wemple-DiDomenico model.

Keywords: Gamma Radiation; PEMA; X-Ray Diffraction Technique; Optical Parameters; Dispersion Parameters; TGA Thermogravimetric

1. INTRODUCTION

Acrylic and methacrylic polymers have characteristics of brilliance, optical clarity, high transparency, improved mechanical properties, adhesion capability and chemical stability [1]. The photostability of aliphatic acrylic and methacrylic polymers are generally very high. Applications for rigid methacrylate polymer products include signs and glazing substitute for safety glass and to prevent vandalism, optical fibers for light transmission, plastic eyeglass lenses, dentures and dental filling materials, and contact lenses (hard and soft) [2]. Polymethacrylates degrade to lower molecular weight compounds or to the monomers on heating or irradiation with high energy radiation. The decomposition or stability of methacrylate polymers is related to their structures such as type of ester groups molecular weight and its distribution, stereoregularity and copolymer composition. Although data have been reported on the glass transition temperature [3-5] of poly(ethyl methacrylate), its thermal degradation has been studied [6,7]. [8] Malhotra et al. however in their isothermal weight loss studies report that up to 40% weight loss values, the thermal decomposition of PEMA is closely resembling that of PMMA. Their GPC analysis also shows that the decomposition of PEMA is dominated by depolymerization. The nature of relaxation mechanisms in amorphous PEMA is discussed on the basis of the TSDC and dielectric results [9].

In general, thermal stability of the polymer composites play a crucial role in determining application, because it effects the final properties of the polymer composites such as the upper-limit use temperature and dimensional stability. For the fabrication of advanced composites with better balance in processing and performance, it is very instructive to characterize the thermal decomposition behavior of the polymer composites. In this regard, thermogravimetric analysis (TGA) is in common use to characterize thermal decomposition behavior and the kinetic parameters of thermal decomposition processes, because of its simplicity and useful information about thermal decomposition [10-14]. Consequently, the understanding of the thermal stability and thermal decomposition kinetics of polymer composites make it possible to develop and extend their applications as various industrial fields. At the same time, the need to develop a molecular-level understanding of the thermal degradation of polymers is becoming increasingly important in areas...
of science and technology associated with the high-
temperature processing and combustion of these mate-
rials.

Studies continue to focus on developing new methods
for increasing the radiation resistance of polymeric mate-
rials since radiation exposure is a concern in many poly-
mers applications. The smaller the extent of mole-
cular changes due to irradiation, the more radiation
resistant the material. Modification of polymers by irra-
diation, with ionizing radiation in particular, is an ex-
panding field of research and application because of its
technological implications. In addition, γ-irradiation has
been proved to be a significant tool [15] for such modi-
fications. It is well known that γ-irradiation can amend
the electrical, optical, thermal properties etc. of the poly-
meric materials by changing their morphology through
various processes of chain accessioning and cross linking
[16-18]. The polymer irradiation leads to shift in optical
absorption edges, which shows a lowering of the en-
ergy-gap. The decrease in energy band-gap infers an in-
crease in conductivity of the irradiated polymers which
might be due to the formation of clusters with rings, or
due to the formation of conjugated double bonds or
quinonic structures [19].

In the present work the affect of different gamma
doses (10 to 100 K Gy) on the optical and thermal de-
composition character of PEMA has been studied by
UV-Vis spectroscopy and TGA measurements. The affect
of irradiation dose on the optical and thermal properties
has been discussed in the light of the polymer’s track
registration sensitivity. The kinetics of decomposition of
PEMA has been studied under non-isothermal conditions
using thermo-gravimetry to determine the activation en-
ergy ($E_a$). Thermal behavioral analysis of PEMA as an
affect of γ-irradiation have attracted much attention be-
cause of their application in optical devices, and has an
important study in this direction.

2. EXPERIMENTAL

2.1. Materials

Poly(ethyl-methacrylate) (PEMA) with a molecular
weight of ~280,000 g mol$^{-1}$ was purchased from Scien-
tific Polymer Products, Inc. (Ontario, NY) and used with-
out further purification.

2.2. Method

The samples for these works were prepared by em-
ploying the solution casting technique. Films of PEMA
were prepared by first dissolving 1 - 2 g of PEMA in
approximately 20 mL of chloroform. The mixtures were
stirred for 24 h using digital magnetic stirrers to ensure
that the PEMA powders fully dissolved. After thoroughly
dissolved, the sample was cast into Petri dish and let to
dry at room temperature for 48 h to ensure complete
evaporation of the chloroform without promoting ther-
mal degradation. The resultant thin PEMA film (0.5 mm)
was carefully removed from the molds for subsequent
analysis.

2.3. Samples Irradiation

Sample irradiation was carried out using a $^{60}$Co gamma
source model ISSLEDOVATEL manufactured by Rus-

sian irradiator and located at NCRRT, Egypt. The dose
rate was 910 kGy/100 min and the temperature during
irradiation was about 40°C. Irradiation doses ranged from
10 to 100 kGy in air.

2.4. Structural Characterization and Thermal
Decomposition

X-ray diffraction (XRD) patterns were recorded using
Shimadzu diffractometer XRD-6000 x-ray diffraction
spectrometer with a copper target (λ = 1.542 Å) at oper-
ating voltage of 40 kV and an electric current of 30 mA.
The optical absorbance and transmittance spectra were
recorded against wavelengths of 200 - 1100 nm using
UV/vis unikon 360 double beam spectrophotometer. Ther-
modgravimetric analysis (TGA) were performed using
Shimadzu-50 instrument (Japan) at heating rate of
10°C/min under flowing nitrogen (20 mL/min) from am-
bient temperature to 500°C. The primary TGA thermo-
grams were used to determine the kinetic parameters
such as activation energy and order of the thermal de-
composition reaction.

3. RESULTS AND DISCUSSION

3.1. X-Ray Diffraction Studies

The X-ray diffraction pattern of (polyethyl-methacry-
late) PEMA samples are shown in Figure 1. The patterns
revealed that there are no sharp diffraction lines and the
broadened background scattering area of PEMA suggests
the presence of amorphous nature, which confirms its
non-crystalline structure. Consequently, the increase in
X-ray intensity may be resulting from grain growth. The
particle consists of grains which are intensity intercon-
ected, causing particles to be almost without porosity.
This appears as boarding peaks. When PEMA films are
irradiated with γ-rays, a cross-linked product is formed.
The product is dimensionally stable and has increased
thermal stability and a significant increase in tensile
strength. So, it is well known that the interaction of ra-
diation with polymer materials result in the formation of
free radicals by dissociation of the excited states or by
ion-molecular reactions.
3.2. Thermogravimetric Analysis Techniques (TGA)

The thermal stability of the polymer composites play a crucial role in determining the limit of their working temperature and the environmental conditions for use, which are related to their thermal decomposition temperature and decomposition rate [20]. To study the affect of γ-irradiation on these kinetic parameters, TGA thermograms of pristine and gamma irradiated samples of PEMA polymer at different doses have been recorded and are presented in Figure 2. TGA thermograms of PEMA polymer at 100°C/min under nitrogen is shown. TGA curves of PEMA polymer shifted toward higher temperature regions with γ-irradiation doses, which was explained by the fact that polymer molecules did not have enough time to exhaust the heat with increasing γ-irradiation doses, leading to slower decomposition rate and higher decomposition temperature due to slow diffusion of heat [21]. TGA provides a method for thermal stability testing. It is observed that there are three stages of decomposition appearing in TGA curves as shown in Figure 2. The first stage at decomposition temperature ≈ 2300°C is attributed to elimination of evaporated molecules in the side groups. The second stage of weight loss in the temperature range ≈230°C - 4000°C is attributed to quaternized graft chain degradation and the third stage of decomposition of the backbone polymer.

TGA thermograms, various kinetic parameters of degradation reaction have been determined by adopting the most commonly used method of Horowitz-Metzger [22-27]. The activation energies of virgin and γ-irradiated samples of PEMA polymer corresponding with the major degradation process have been deduced using the expression [27].

\[
\ln\left(\ln\left(\frac{W_0}{W_f} \div \frac{W_r}{W_f}\right)\right) = \frac{E_a}{RT} \times \theta
\]

where \(W_0\), \(W_f\) are the initial and final weights, \(W_r\) is the remaining weight at temperature \(T\), \(E_a\) is the activation energy, \(R\) is gas constant (\(R = 8.314 \text{ J·K}^{-1}·\text{mol}^{-1}\)), and \(\theta = T - T_s\) with \(T_s\) as the reference temperature corresponding to \(W_r/\). In the light of the Eq.1, the activation energy \(E_a\) can be calculated from the slope of the linear fitted line between \(\ln\left(\ln\left(\frac{W_0}{W_f} \div \frac{W_r}{W_f}\right)\right)\) and \(\theta\) as illustrated in Figure 3, for virgin and γ-irradiated samples at doses 10, 20, 40, 60 and 100 kGy. The values of activation energies so determined for pristine as well as γ-irradiated samples have been enlisted in Table 1. There is a clear cut increase in the values of activation energy with the increasing irradiation dose. Such an increase may be attributed to the initialization of chain scissioning, possible evaporation of volatile side groups resulting in significant reduction of packing density, reorganization of molecular arrangements etc. in the polymeric sample which signifies the decrease in thermal stability of the polymer [22,23,28].

3.3. Optical Studies

3.3.1. The Optical Absorption Edge

The study of the optical absorption and particularly the absorption edge is an useful method for the investigation of optically induced transitions and for the provision of information about the structure and energy gap in both crystalline and non-crystalline materials. The principle of this technique is that a photon with energies greater than the band gap energy will be absorbed. The absorption coefficient \(\alpha(\omega)\) depends exponentially on the photon energy \(h\omega\). The absorption edge in many disordered materials follows the Urbach (1953) rule [29] given by:

\[
\alpha(\omega) = \beta \exp\left(\frac{h\omega}{\Delta E}\right)
\]

where \(\alpha(\omega)\) is the absorption coefficient at an angular...
frequency of $\omega = 2\pi \nu$ and $\beta$ is a constant. $\Delta E$ is the width of tail of localized states in the band gap. The lower energy part of the absorption spectra gives information about the vibrations, and the higher part of the spectrum gives information about electronic states in the normal material. Figure 4 shows the absorption characteristic for the un-irradiated and irradiated EPMA Films. It is clear from Figure 4 that blank shows one absorption bands. The absorption at 325 nm is identified as the soreset band (B-band). These bands are attributed to the carbonyl group in polymeric macromolecule. It is observed that dose dependence UV has a slight effect on the position and the intensity of this band for PEMA films. Moreover, the intensity of carbonyl group peak decreases by increasing irradiation doses whereas its position is slightly shifted for all films.

3.3.2. The Optical Band Gap

A model based on the electronic transition between the localized states is not preferable. For higher values ($\alpha \geq 10^5$ cm$^{-1}$), the absorption coefficient (where absorption is associated with inter-band transitions) [30,31] takes the form:

$$\alpha(\omega) = \beta(h\omega - E_{opt})^M / h\omega$$

$E_{opt}$ the optical gap, and a number that characterizes the transition process. The value of the constant $\beta$ in Eq.3 can be evaluated from the slope of the straight part in Figure 4. In the high absorption region where absorption is associated with inter-band transitions, there are two kinds of optical transition at the fundamental edge of crystalline and non-crystalline, direct transitions and indirect transitions. Both of which involve the interaction of an electromagnetic wave with an electron in valance band, which is then raised across the fundamental gap to the conduction band.

For indirect transitions, interactions with lattice vibrations (phonons) take place; thus the wave vector of the
electron can change in the optical transition and momentum change will be taken or given up by phonons. In other words, if the minimum of the conduction band lies in a different part of K-space from the maximum of the valence band, a direct optical transition from the top of the valence band to the bottom of the conduction band is forbidden.

For direct transition $M = 1/2$ or $3/2$ while for indirect transition $M = 1, 2$ and 3 which characterizes many amorphous or semi crystalline substances. The changes are strongly dependent on the internal structure of the absorbed substances. It may be expected that the gamma radiation can cause ionization or excitation of the optical electrons and possibly, displacement of atoms from their sites in the lattice of solid. When PEMA films are irradiated with radiation cross-linking and chain scission are formed. The optical band gap, $E_{opt}$ for un-irradiated and irradiated samples was determined using Eq.3, from the intercepts of the extrapolated linear part of plots of $(\alpha h \nu)^{1/2}$ versus $(h \nu)$, as shown in Figure 5. The values of $E_{opt}$ obtained are listed in Table 2 and it was found that $E_{opt}$ decreases with increasing of the irradiation dose.

### 3.3.3. The Band Tail Width

The band tail width was obtained using Eq.3 and then by plotting $(\ln \alpha)$ against $(h \nu)$ as shown in Figure 6, that should yield a straight line of a slope equals $(1/\Delta E)$. It should mention that Eq.3 is applied only in the low absorption region. The band tail width $E_e$ is listed in Table 2, it increases with increasing dose; i.e. $E_e$ exhibits an opposite trend with $E_{opt}$. This is could be due to the fact that, when polymer exposed to irradiation doses, it starts to crosslink, a formation of new covalent bonds can be produced. Then, different chains are obtained, i.e. cross...
linking starts which are in turn hinder the motion of molecules and reduce its activities and in turn decrease the optical band gap. On further gamma doses, a formation of free radicals are obtained, which are causes a decrease in the optical band gap. As the gamma dose increases from 10 to 100 kGy, the crystalline structure is assumed to be perturbed and leading to an increase in the degree of disorder. At higher dose (100 kGy), the chains scission become predominant relative to the crosslinking. From the density of states model it is known that $E_{\text{opt}}$ decreases with increasing the degree of disorder of amorphous phase. Furthermore, the band tailing shifts to higher energies as shown in Figure 5 and extends into the forbidden band.

### 3.3.4. Dispersion Analysis

In this work, the polymer films are sufficiently thick compared to the beam wavelength to neglect the interference phenomena. The air/polymer interface transmission $T$ are expressed as functions of refractive index ($n$) and extinction coefficient ($K$) [32].

\[
T = \frac{4n}{(n+1)^2 + K^2} \tag{4}
\]

Once the measurements $T$ have been performed, an iterative procedure to calculate $n$ and $K$ is initiated. When the absorption $\alpha$ is known, the extinction coefficient $K$ can be found the relation $K = \alpha \lambda / 4\pi$. The real and
imaginary parts of dielectric constant and can be estimated if the refractive index and extinction coefficient is known from the relations $\varepsilon' = n^2 - K^2$ and $\varepsilon'' = 2nK$. The real part generally relates to dispersion, while the imaginary part provides a measure to the dissipative rate of the wave in medium [33].

The refractive index and the single oscillation parameters have been calculated and discussed in terms of the Wemple-DiDmenico model [34] Figure 7. That the parameters single-oscillation energy ($E_0$) and dispersion energy ($E_d$) have been calculated by knowing the refractive index ($n$) which is independent on the scale of optical dielectric constant ($\varepsilon$) and is consequently an “average” energy gap, where as $E_d$ is depends on the scale of $\varepsilon$ and thus serves as an interband strength parameters. Since the $M_o$ and $M_d$ moments are involved most heavily near the interband absorption edge [35].

The dispersion of refractive index below the interband absorption edge according to the single oscillator model is given by [34].

$$n^2(E) = 1 + \frac{E_0 E_d}{E_0^2 - E^2}$$

(5)

where the parameters $E_0$, $E_d$ are the single-oscillation energy and dispersion energy respectively. By plotting $(n^2 - 1)^{-1}$ versus $E^2$ and fitting a straight line as shown in Figure 8; $E_0$ and $E_d$ are determined directly from the gradient, $(E_0, E_d)^{-1}$ and the intercept $(E_0/E_d)$, on the vertical axis, Table 3 and Figure 9 [35-37].

The XRD data suggested that the crosslinking process predominated during the irradiation of PEMA. 2) TGA studies revealed that the thermal stability of polyethyl methacrylate.

3) Gamma-rays irradiation of PEMA films in the dose range of 10 up to 100 kGy has been investigated as a potential technique of refractive index modulating opti-

Figure 7. Refractive index ($n$) against wavelength ($\lambda$) nm, for PEMA film before and after irradiation with different dose.

4. CONCLUSIONS

From the results and discussion above it is possible to conclude that:

1) XRD obtained for PEMA films confirmed the amorphous nature and partially crystalline nature for different films. The XRD data suggested that the crosslinking process predominated during the irradiation of PEMA.

2) TGA studies revealed that the thermal stability of polyethyl methacrylate.

3) Gamma-rays irradiation of PEMA films in the dose range of 10 up to 100 kGy has been investigated as a potential technique of refractive index modulating opti-
506
cal elements. There was a decrease in the refractive index in via the decrease in the dispersion energy for the irradiated films with different doses.

4) Single oscillator energies in the PEMA films were significantly influenced by gamma irradiation affects. It was observed a decrease in the dispersion energy and discussed in terms of the Wemple-DiDomenico model.

5) The affect of gamma irradiation on the optical band gap obtained from. Moreover, there was a decrease in the optical energy gap with increasing radiation dose up to 100 kGy. It was deduced that the energy of gamma rays, in the dose range from 10 up to 100 kGy, is suitable for the cross-linking of PEMA films.

REFERENCES

[1] Paul, S. (1996) Surface coatings science and technology. Wiley, New York, 312.

[2] Odian, G. (1991) Principles of polymerization. John Wiley & Sons Inc., New York, 311-313.

[3] Shetter, J.A. (1963) Effect of stereoregularity on the glass temperatures of a series of polyacrylates and polymethacrylates. *Journal of Polymer Science, B,* 1, 209-213. doi:10.1002/pol.1963.110010503

[4] Hata, T., Nose, T. and Polym, J. (1967) Stress-strain-time relation of poly(n-alkyl methacrylates) in rubbery state. *Science, C,* 16, 209.

[5] Gardlund, Z.G., Laverty, J.J. and Polym, J. (1969) Polyalkylmethacrylate films as matrices for photochromic studies. *Science, B,* 7, 719.

[6] Grassie, N., Mc Callum, J.R. and Polym, J. (1964) Thermal and photochemical degradation of poly(n-butyl methacrylate). *Science, A,* 2, 983.

[7] Smith, S.D., Long, T.E. Mc Grath, J.E. and Polym, J. (1994) Thermogravimetric analysis of poly(alkyl methacrylates) and poly(methylmethacrylate-g-dimethyl siloxane) graft copolymers. *Science-Chemistry A,* 32, 1747.

[8] Malhotra, S.L., Minh, L., Blanchard, L.P. and Macromol, J. (1983) Thermal decomposition and glass transition temperature of poly-(phenylmethacrylate) and poly (cyclohexyl methacrylate). *Science-Chemistry A,* 19, 559.

[9] Ahmed, M.T. (2010) Relaxational study of poly(ethylmethacrylate) by thermally stimulated depolarization current, thermal sampling spectroscopy: Modified dipole-dipole interaction theory. *Journal of the Korean Physical Society,* 57, 272-281. doi:10.3938/jkps.57.272

[10] Maitra, S., Bandypadhyay, N., Das, S., Pal, A.J. and Pramanik, J. (2007) Non-isothermal decomposition kinetics of alkaline earth metal carbonates. *Journal of the American Ceramic Society,* 90, 1299-1303. doi:10.1111/j.1551-2916.2007.01607.x

[11] Friedman, H.L. (1964). Kinetic of thermal degradation of char-forming plastics from thermogravimetry. Application to a phenolic plastic. *Journal of Polymer Science Polymer Symposium,* 6, 183-195. doi:10.1002/polc.5070060121

[12] Nam, J.D. and Seferis, J.C. (1991) Composite methodology for multistage degradation of polymer. *Journal of Polymer Science: Part B, Polymer Physics,* 29, 601-608. doi:10.1002/polb.1991.000290509

[13] Anderson, D.A. and Freeman, E.S. (1961) The kinetics of thermal degradation of polyethylene and polystyrene. *Journal of Polymer Science,* 54, 253-260. doi:10.1002/pol.1961.1205415920

[14] Radhakrishnan, T.S. (1999) New method for evaluation of kinetic parameters and mechanism of degradation from pyrolysis—GC studies: Thermal degradation of polydimethylsiloxanes. *Journal of Applied Polymer Science,* 73, 441-450. doi:10.1002/(SICI)1097-4628(19990718)73:3<441::AID-APP16>3.0.CO;2-T

[15] Fink, D., Ed. (2004) Fundamentals of ion-irradiated polymers. Springer-Verlag, Berlin, 7-84.

[16] Sharma, T., Aggarwal, S., Kumar, S., Mittal, V.K., Kalsi, P.C. and Manchanda, V.K. (2007) Effect of gamma irradiation on the optical properties of CR-39 polymer. *Journal of Materials Science,* 42, 1127-1130. doi:10.1007/s10853-006-0516-7

[17] Abiona, A.A. and Osinkolu A.G. (2010) Gamma-irradiation induced property modification of polypropylene. *International Journal of Physical Science,* 5, 960-967.

[18] Zaki, M.F. (2008) Gamma-induced modification on optical band gap of CR-39 SSNTD. *Brazilian Journal of Physics,* 38, 558-562. doi:10.1590/S0103-97332008000500005

[19] Fink, D., Chung, W.H., Klett, R., Schmoldt, A., Cardoso, J., Montiel, R., Vazquez, M., Wang, L., Hosoi, F., Omichi, H. and Goppelt-Langer, P. (1995) Carbonaceous Clusters in irradiated polymers as revealed by UV-Vis spectrometry. *Radiation Effects and Defects in Solids,* 133, 193-208. doi:10.1080/016205695094223990

[20] Nair, C.P.R., Bindu, R.L. and Joseph, V.C. (1995) Cyanate esters based on cardanol modified-phenol-formaldehyde resins: Syntheses and thermal characteristics. *Journal of Polymer Science, Part A: Polymer Chemistry,* 33, 621. doi:10.1002/pola.1995.080330403

[21] Tang, W., Li, X.G. and Yan, D. (2004) Thermal decomposition kinetics of thermotropic copolyesters made from trans-p-hydroxyccinnamic acid and p-hydroxybenzoic acid. *Journal of Applied Polymer Science,* 91, 445-454. doi:10.1002/app.13103

[22] Nouh, S.A., Atta, M.R. and El-Melleegy Rad, W.M. (2004) A study of the effects of gamma and laser irradiation on the thermal, optical and structural properties of CR-39 nuclear track detector. *Radiation Effects and Defects in Solids,* 159, 461-474. doi:10.1080/1042150412331296835

[23] Kalsi, P.C., Mudher, K.D.S., Pandey, A.K. and Iyer, R.H. (1995) Thermal studies on unirradiated and γ-irradiated polymer of allyl diglycol carbonate. *Thermochimica Acta,* 254, 331-336. doi:10.1016/0040-6031(94)19839-T

[24] Mallikarjun, K.G. (2004) Thermal decomposition kinetics of Ni(II) chelates of substituted chalcones. *E-Journal of Chemistry,* 1, 105-109. doi:10.1155/2004/385034
[25] Jellinek, H.H.G. (1978) Aspects of degradation and stabilization of polymers. Elsevier Scientific Publishing Company, New York.

[26] Singh, B.K., Kumari, P., Prakash, A. and Adhikari, D. (2009) Thermal decomposition kinetics of peanut shell. Nature and Science, 7, 73-78.

[27] Horowitz, H.H. and Metzger G. (1963) A new analysis of thermogravimetric traces. Analytical Chemistry, 35, 1464-1468. doi:10.1021/ac60203a013

[28] Anslyn, E.V. and Dougherty, D.A. (2006) Modern Physical organic chemistry. Edwards Brothers, Inc., Violet Hill.

[29] Urbach, F. (1953) The long-wavelength edge of photographic sensitivity and of the electronic absorption of solids. Journal of Physical Reviews, 92, 1324. doi:10.1103/PhysRev.92.1324

[30] Ilyas, M., Zulfequar, M. and Husain, M. (2000) Optical properties of a-(Se70Te30)100-x (Se98 Bi2)x thin films. Journal of Modern Optics, 47, 663.

[31] Mohamed, S.H., Ahmed, A.M., Diab, A.K. and Wakkad, M.M. (1999) Structural and optical properties of Ge-As-Te thin films. The European Physical Applied Journal, 8, 215-224.

[32] Soldera, A. and Monterrat, E. (2002) Mid-infrared optical properties of a polymer film: Comparison between classical molecular simulations, spectrometry, and ellipsometry techniques. Polymer, 43, 6027-6035. doi:10.1016/S0032-3861(02)00467-6

[33] Rajesh, K.R. and Menon, C.S. (2002) Estimation of the refractive index and dielectric constants of magnesium phthalocyanine thin films from its optical studies. Materials Letters, 53, 329-332. doi:10.1016/S0167-577X(01)00502-X

[34] Wemple, S.H. and DiDomenico, W. (1971) Behavior of the electronic dielectric constant in covalent and ionic materials. Physical Review B, 3, 1338-1351. doi:10.1103/PhysRevB.3.1338

[35] Dixon, J.R. and Riedl, H.R. (1965) Optical dispersion of lead sulfide in the infrared. Journal of Physical Review A, 140, 1283-1291. doi:10.1103/PhysRev.140.A1283

[36] Sokkar, T.Z.N., El-Farahary, K.A. and El-Bakary M.A. (2003) Determination of optical properties, dispersion, and structural parameters of poly(ethylene terephthalate) fibers using automatic variable wavelength interferometry technique. Journal of Applied Polymer Science, 89, 1737. doi:10.1002/app.12189

[37] Abul-Hail, R.Ch. (2010) Optical absorption of polycarbonate (Makrofol E) as means of gamma-ray dosimetry. Journal of Basrah Researches (Sciences), 36, 26.