Electricity Generation using Carboxymethyl Cellulose and Kitchen Waste as Substrate by *Exiguobacterium* sp SU-5 in Mediatorless Microbial Fuel Cell

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Abstract

Microbial fuel cell (MFC) has become a great attraction amongst most researchers, where degradation of waste takes place simultaneously produces electricity. Using an efficient organism and a better proton exchange membrane gives out good electricity. In this study, *Exiguobacterium* sp SU-5 was isolated from soil and used for producing electricity against carboxy methyl cellulose (CMC) in Nafion membrane and Salt bridge fitted MFC, where both act as proton exchange membrane. The organism was found to produce more electricity in Nafion membrane fitted MFC. Later the organism was subjected to produce electricity against kitchen waste and the kitchen waste was also checked for BOD, COD and other water analysis before and after the treatment. The organism could produce more electricity in Nafion membrane fitted MFC and found to reduce chloride, fluoride and hardness of water.

Keywords: MFC; carboxy methyl cellulose; kitchen waste; electricity.

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INTRODUCTION

Microbial fuel cells (MFCs) have gained the attention of many researchers in recent years which is expected to become a promising means of organic waste management with the production of electrical energy. In MFC, the interaction between the microorganisms and the electrodes is brought by electrons transfer which further completes the electrical circuit to produce current. MFCs are bioelectrochemical systems (BESs) which generates electricity through microbial metabolic degradation of the feed biomass. The efficacy of MFCs is dependent on its design and construction, recently researchers have studied the performance of MFCs with varied designs and unit components to increase its efficiency. In a single chambered MFC, the microorganisms oxidize the organic matter in the anode chamber (Anaerobic conditions) and releases protons and electrons. These free electrons travel through the external circuit to reach the cathode which is directly exposed to air. While in a two chambered MFC, the proton transfer to the cathode occurs through ion-selective membrane. This prevents oxygen diffusion to anode and the anaerobic condition at anode chamber is well preserved. At the cathode (usually aerobic), the protons and electron acceptor (mainly oxygen) combine to produce water and the flow of electrons creates electrical output. Besides these two common designs, several modifications have been incorporated in MFC design and structure.

In MFCs, the material choice for substrate is considered as the most important biological factor affecting the electric output. A great variety of substrates present in wastewater either in its pure compounds or as complex composition of organic matters can be used in MFCs for electricity production. Acetate is a simple carbon source and is widely used to promote the action of electroactive bacteria in MFC. It is also the end product of several metabolic pathways for higher order carbon sources which includes Entner–Doudoroff pathway for glucose metabolism. Another commonly used substrate compound in MFCs is glucose. A maximum power density of 216 W/m² was recorded from a glucose fed-batch MFC using 100mM ferric cyanide as cathode oxidant. Also, the substrate degradation by multispecies microbial community in MFCs show better energy production and substrate conversion. In a two-chambered MFCs with fed-batch control system using ferricyanide as the catholyte, the co-culture of the electrochemically active Geobacter sulfurreducens and the cellulolytic fermenter Clostridium cellulolyticum achieved maximum power densities of 59.2mW/m² and 143mW/m² (anode area) from 1g/L carboxymethyl cellulose (CMC) and MN301 cellulose, respectively. Neither of the pure culture produced electricity from these substrates when used alone. The yield from CMC degradation by co-culturing increased from 42% to 64% compared to pure C.cellulolyticum culture. Likewise, rumen bacterial consortia were found to produce electricity with a power density of 55 mW/m² by consuming cellulose. Huang and Logan experimented on paper recycling plant wastewater by adding 50mM phosphate buffer solution (PBS, 5.9 mS/cm) which resulted with the power generation of 501± 20 mW/m²; with a coulombic efficiency of 16±2%, with 73± 1% removal of Soluble organic matter, 96±1% removal of cellulose over a 500-h batch cycle. On evaluating the potential of various eco-systems through benthic fuel cells, it was attributed that water bodies with high organic matter could account for large power output. Mohan et al. also stated a similar report where the substrate concentration (carbon source) in the water body showed significant influence on bioelectricity production in benthic fuel cell since they act as potent electron donors. Thus, Water bodies with higher organic matter can also serve as better substrate for MFCs.

Therefore, it is clear that the choice of substrate and the acting microbial consortia reflects in the magnitude of current produced. Here, cellulose degrading organisms was isolated from soil sample and their efficiency of electricity production based on the cellulose degradation by single organism was studied in two different MFC models, one fitted with Nafion membrane and the other consists of a salt bridged system. 0.5% and 1% CMC (Carboxy Methyl Cellulose) was used as the substrate material for the isolated organism and the energy produced in each case, Nafion MFC and Salt bridge MFC was recorded. The electricity generated by the single organism on a single substrate was compared to the electricity produced by multiple consortia present in organic waste matter, like the kitchen waste. Kitchen
waste acting as an enriched source of necessary metabolic components which is predominantly of plant-based products serve as a platform consisting of various active micro-organisms. MFC was set up using the kitchen waste where the microbial degradation by instinctive microbes were converted into energy production. Construction of MFC using the Kitchen waste can act as the best means of waste management with simultaneous production of electricity.

MATERIALS AND METHODS
Screening, Isolation and Identification of Microorganism
The microbial samples were collected from the garden soil of Sathyabama Institute of Science and Technology, Chennai. A total of 4 soil samples were taken in sterile containers. 1 g of each sample was inoculated into tube containing minimal media (Potassium dihydrogen phosphate (15g/L), Ammonium chloride (5g/L), Disodium hydrogen phosphate (64g/L), Sodium chloride (2.5g/L), 1M Magnesium sulphate (2mL/L), 1M Calcium chloride (0.1mL/L)) with 1% of carboxy methyl cellulose (CMC) and incubated at room temperature. After 24 h of incubation, the viscosity of the medium was compared with the control. After 48 h of incubation, the media was serially diluted and subject for pour plate using 1% CMC containing minimal media agar. The colonies grown on cellulose medium was isolated and used for the study. Identification of the organism was done by biochemical tests and 16S rRNA sequencing\textsuperscript{11}.

Collection of Kitchen Waste
The Kitchen waste was collected from boy’s mess, Sathyabama Institute of Science and Technology, Chennai. It was used as the source material for microbial degradation and thereby produce electricity.

Construction of Microbial Fuel Cell
Pretreatment of Nafion Membrane
The purchased Nafion membrane 117 (3*3 cm, 50 pore size) was pretreated by heating at 80°C using 5% aqueous Hydrogen peroxide (H\textsubscript{2}O\textsubscript{2}, 30% in H\textsubscript{2}O) for 2 h, rinsed 3-4 times in distilled water then placed in boiling water for 2 h and finally washed in deionized water. Finally fixed to the MFC and checked for leakage\textsuperscript{12}.

Pretreatment of Electrode
The graphite electrodes were soaked in IPA (Iso propyl Alcohol) for 30 mins and washed with 1N HCl and 1N NaOH to eliminate the fouling of metal\textsuperscript{13}.

Cathode Chamber
10mM of Potassium permanganate (KMnO\textsubscript{4}) was aerated with mechanical pumps, which is used as the catholyte\textsuperscript{14}.

Fig. 1. Microbial fuel cell – double chambered fuel cell (with nafion membrane) a) anode chamber, b) cathode chamber c) multimeter.

Fig. 2. Microbial fuel cell – double chambered fuel cell (with salt bridge) a) anode chamber, b) cathode chamber c) multimeter.
Construction of Microbial Fuel Cell and Electricity generation in MFC

The H-shaped microbial fuel cell comprises of an aerated cathode chamber and an anaerobic anode. Both the chambers were connected using a clamp which is fitted with proton exchange membrane (Nafion™ DuPont) Fig. 1. The carbon electrode of size 69.5325 cm² were connected by means of an insulated copper wire to complete the circuit. Anode chamber was having the capacity of 375 mL. The anaerobic condition at the anode was configured by sealing with PTFE (Poly tetra fluoro ethylene) tapes and aeration was provided to the cathode using the air pump Fig. 1.

Another MFC was constructed with similar components such as anaerobic anode chamber and an aerobic cathode chamber Fig. 2 but the connection between the two chambers was brought with salt bridge as the proton exchange of length 11.5 cm. The salt bridged MFC were constructed using available plastic bottles of capacity 500 mL, which were connected using salt bridge made of PVC pipe filled with saturated NaCl in 2% agar (boiled and poured). Graphite electrodes of size 69.5325 cm² was used in both the chambers. Anaerobic condition was maintained in the anode chamber whereas aerobic condition was maintained in cathode chamber using aerator pumps. Both the setups were kept in room temperature 28 ± 2°C.

The anode compartment of constructed units were loaded with 375 ml for Nafion membrane chamber and 500 ml for the salt bridge setup with either 0.5% or 1% CMC in minimal media (Potassium dihydrogen phosphate (15g/L), Ammonium chloride (5g/L), Disodium hydrogen phosphate (64g/L), Sodium chloride (2.5g/L), 1M Magnesium sulphate (2mL/L), 1M Calcium chloride (0.1mL/L)) and inoculated with the isolated...
bacterial culture (1% inoculum). The microbial activity in the anode chamber was preserved at anaerobic condition throughout the study. Using a digital multimeter, the current flow and the potential difference between the electrodes in MFC were measured at regular time interval for 10 to 11 days only. Later, the source material of anode chamber was replaced with the kitchen waste collected from the Boy’s mess, Sathyabama Institute of Science and Technology. Prior to adding to the anode chamber, kitchen waste was ground to semisolid state, made up to 200 mL and autoclaved.

**Monitoring the Readings**

The current/voltage generated by the constructed MFC’s were measured using a rheostat by constant current and constant voltage method.

**Analysis of Kitchen Waste**

Kitchen wastewater was subjected for following analysis before and after (10 days) subjugating to MFC: protein content, carbohydrate content, BOD / COD analysis\(^{18,19}\), Estimation of Chloride, Fluoride, Nitrate, Iron Residual Free Chlorine Test and Hardness of Water was analysed using water testing kit (HiMedia, WT023)

**RESULTS**

**Screening, Isolation and Identification of Isolated Microorganism**

The sample was found to liquify the cellulose media and was subjected for isolation and identification. One organism was isolated and identified on the basis of 16SrRNA sequencing as *Exiguobacterium* SU-5 (GENBANK accession number GU395989).

**Generation of Electricity from Microbial Fuel Cell MFC with 0.5% and 1% CMC Media**

The constructed units were loaded with 0.5% and 1% CMC media inoculated with the isolated colony, *Exiguobacterium* sp SU-5. From the graph Fig.3 it is observed that when 0.5% CMC media was used, the electricity production was greater in Nafion membrane fitted Microbial fuel cell compared to salt bridged microbial fuel cell. Nafion membrane fitted Microbial fuel cell showed an upsurge of electricity production during the initial hours of incubation and reached a maximum production of 1067 mV current at the 18\(^{th}\) hour. On further incubation, the electricity generation dropped gradually reaching a constant production of 800 mV from the 44\(^{th}\) h. The prolonged incubation does not seem to have influenced the power generation where the minimum production

| S.No | Parameter          | Concentration (mg/L) |
|------|--------------------|-----------------------|
| 1    | Protein Estimation | 137.5                 |
| 2    | Reducing sugar     | 124                   |
| 3    | COD                | 180                   |
| 4    | BOD                | 19.5                  |
| 5    | Chloride Test      | 150                   |
| 6    | Fluoride Test      | 2.0                   |
| 7    | Nitrate Test       | 0.0                   |
| 8    | Iron Test          | 0.5                   |
| 9    | RF Chlorine Test   | 0.0                   |
| 10   | Hardness Of Water  | 1975                  |

**Table 1. Analysis of kitchen waste treatment in MFC**

**Fig. 5. Electricity generation utilizing kitchen waste by *Exiguobacterium* sp SU-5.**
of 800 mV was maintained on incubation beyond 44th h. In salt bridge system, the initial hours of incubation developed a constant voltage of 200mV till 92nd h and an increase in current, producing 580mV which is relatively less compared to Nafion MFC. This was concluded with the fact that the efficiency of proton transfer through salt bridge from anodic chamber to cathodic chamber was minimal than the Nafion membrane. In 1% CMC Media MFC, *Exiguobacterium* sp SU-5 generated 1350 – 1400 mV in Nafion fitted MFC unit from the 140th h and continued till the 250th h whereas only around 500 mV of electricity was produced in salt bridge system from 48th hour and then it got reduced at 168th hour and did not get increased (Fig. 4). Again, Nafion MFC was found to be the best system for producing electricity while degrading cellulose.

Kitchen waste collected from boy’s mess, Sathyabama Institute of Science and Technology was found to produce appreciable quantity of electricity than ever reported i.e it constantly produced electricity of 1400 mV upto 170 hours Fig.5. Again, Salt bridge system failed to produce more electricity as it produced in Nafion MFC. It was reported that the salt concentration used in the salt bridge can affect the channelling of protons. An increased molar concentration can account for high internal resistance and thereby produce lower current20. To be more analogous with the above result, Shahi et al has also reported the same using wastewater as a substrate. He performed a comparative study with Nafion and agar salt bridged MFCs where the power output from MFCs constructed with agar salt bridge (0.145 V) was much lower to Nafion membrane fitted MFC (0.504 V). Therefore, it clearly explains that the membrane offers less resistance to proton channelling when compared with salt bridge21. In a mediator less microbial fuel cell, electron transfer between a microbe and an electrode takes place through the active centre of the released enzymes which directly get allied to the electrode thus facilitates electron transfer. In such a case, the rate of electron transfer can be very low if the active site of the involving enzymes is insulated by random proteins in the environment and the isolation of the enzyme from the electrode surface by its relative burial into the bacterial membrane. Whereas in exoelectrogens species, the redox enzymes are present on the outer surface of the microbial membrane which participate in electron transport. Electron transfer can also be active through pilli of 2-3mm long which is also referred as biological nanowires22.

**Analysis of kitchen waste**

The protein content was assessed from kitchen waste by Lowry’s method23,24. The protein quantity reduced from 137.5 mg/l to 112.5 mg/l after treatment. The reducing sugar were quantified by DNS method where an initial concentration of 124 mg/l were utilized to reach a final concentration of 23 mg/l after treatment (Table 1). It is evident that the organism had a significant effect on the other parameters too. It was found to reduce the hardness of water, chloride content and fluoride content (Table 1). Chitosan and wastewater had been used for production of bioelectricity in MFC25,26.

**CONCLUSION**

*Exiguobacterium* sp SU-5 was collected from the garden soil and utilized for electricity generation from Carboxymethyl cellulose and kitchen waste in Nafion membrane and Salt bridge fitted mediatorless MFC. More electricity was found to be produced in Nafion membrane fitted MFC.

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**CONFLICT OF INTEREST**

The authors declare that there is no conflict of interest.

**AUTHOR’S CONTRIBUTION**

All the authors involved in idea creation and made intellectual contribution to the work. All the authors involved in manuscript preparation.

**FUNDING**

None.

**DATA AVAILABILITY**

Not applicable.
ETHICS STATEMENT

No animals or humans are used in this study. Hence, the work does not require ethical clearance.

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