Abstract: Rice sheath blight, a fungal disease caused by *Rhizoctonia solani*, seriously threatens rice production. Some of the volatile organic compounds (VOCs) produced by microbes are inhibitory to the growth of the plant pathogen, and hence may have the potential as environmentally friendly antifungal substances. However, information on the inhibitory effect of VOCs released by rice rhizosphere bacteria on *R. solani* is scarce. In this study, bacteria from the rice rhizosphere capable of inhibiting the growth of *R. solani* via releasing VOCs were screened using a double Petri dish assay. Headspace solid phase microextraction and gas chromatography mass spectrometry (GC/MS) were used to identify and quantify the VOCs. The contributions of VOCs to the inhibition of the growth of *R. solani* were estimated by constructing a random forest model, and were verified using pure compounds. Nine strains (i.e., *Pseudomonas* sp. No. 3, *Enterobacter* sp. No. 26, *Enterobacter* sp. No. 34, *Pseudomonas* sp. No. 35, *Ralstonia* sp. No. 50, *Bacillus* sp. No. 62, *Arthrobacter* sp. No. 146, *Brevibacillus* sp. No. 2–18, and *Paenibacillus* sp. No. 2–60) showed various inhibition on the growth of *R. solani* via VOCs. The inhibitory effect ranged from 7.84% to 100%, with *Ralstonia* sp. No. 50 completely inhibiting the growth of *R. solani*. Five VOCs (i.e., benzoic acid ethyl ester, 3-methylbutanoic acid, 2-ethyl-1-hexanol, 3-methyl-1-butanol, and 6-methyl-5-hepten-2-one) identified by random forest model were confirmed to be toxic to *R. solani* when applied as a pure chemical compound. In particular, benzoic acid ethyl ester, 3-methylbutanoic acid, and 2-ethyl-1-hexanol were lethal to *R. solani*. In summary, the rice rhizosphere bacteria (*Ralstonia* sp. No. 50) and VOCs (benzoic acid ethyl ester, 3-methylbutanoic acid, and 2-ethyl-1-hexanol) showed potential to be used as new resources for biological control of rice sheath blight.

Keywords: rice sheath blight; *Rhizoctonia solani*; volatile organic compounds; random forest model; inhibitory

1. Introduction

Rice is an important staple food crop globally. More than 3.5 billion people rely on it to provide calorific energy every day. Rice sheath blight caused by *Rhizoctonia solani* is one of the most economically important rice diseases worldwide, causing devastating crop losses and representing a serious threat to global food security [1,2]. Currently, rice sheath blight is mainly controlled by fungicides [3]. For instance, Jingangmycin has been used widely to control sheath blight in China for the past 30 years [4]. However, the extensive and continuous use of a single fungicide increases the risk of resistance development in the pathogen [5]. Cultivating resistant rice varieties would be the best option to cope with the...
attack of this pathogen, but no such variety is available to the growers at present [6]. A range of alternate control measures such as fertilization, soil amendments, and agricultural management have also been recommended. Unfortunately, these alternative measures are time-consuming and laborious, which is not particularly effective [7–9]. Therefore, it is imperative to search for new active agents against rice sheath blight.

Biological control is widely studied as an environmentally friendly control strategy. For example, Kanjanamaneesathian et al. found three bacterial isolates that provided maximum suppression of sheath blight lesions [10]. *Bacillus subtilis* B4 is also shown to be a highly effective inhibitor of *R. solani* mycelial growth [11]. During rice growth, the root system can release various signal molecules to shape a unique root microbial community [12]. This microbial community plays an important role in rice growth and development. By releasing various metabolites, such as phytohormones and volatile organic compounds (VOCs), they not only promote plant growth but also help plants resist various diseases [13,14]. So far, there are a few studies on rice-associated bacteria inhibiting the growth of *R. solani* [2], but no reports on whether rice rhizosphere bacteria can inhibit *R. solani* growth via VOCs.

Among the many biological control strategies, VOCs have been studied widely [15,16]. They are potentially effective being small organic molecules (<C20) with low water solubility and high vapor pressure (0.01 kPa at 20 °C) that can readily evaporate and diffuse through heterogeneous mixtures of solids, liquids, and gasses. There is growing evidence that VOCs could modulate various biotic and abiotic stresses in plants, and are very likely to become a substitute for environmentally harmful pesticides and fungicides [17]. For instance, the endophytic bacteria *Pseudomonas stutzeri* E25 and *Stenotrophomonas maltophilia* CR71 inhibited the growth of *Botrytis cinerea* by releasing VOCs, with dimethyl disulfide (DMDS) being the main component [18]. VOCs produced by *Bacillus subtilis* GB03 can directly inhibit the mycelial growth of Arabidopsis gray mold and interfere with the attachment of pathogenic fungi on hydrophobic foliage [19]. A small number of VOCs were found to inhibit growth of *R. solani* specifically, such as sulfide, 1-methyl-4-(1-methylethenyl)cyclohexene, methyl-2-methylvalerate, and 1,3,5-trichloro-2-methoxybenzene [14,19,20]. Overall, information on the identity and inhibitory effect of VOCs released by rice rhizosphere bacteria is fragmented.

In the present study, we searched for novel VOCs from rice rhizosphere bacteria with efficacy against rice sheath blight. To achieve this objective, bacteria from rice rhizosphere suppressive to *R. solani* were screened via a double Petri dish assay, VOCs were identified and quantified with GC-MS, and relative contribution of each VOC was estimated by constructing a random forest model and was further verified using pure compounds.

2. Materials and Methods

2.1. Test Strains, Culture Medium, and Culture Conditions

The test strains were isolated from the rice rhizosphere. All strains were sequenced at the Institute of Crop Science, Chinese Academy of Agricultural Sciences, and then the bacterial species were identified by BLAST search of 16S rDNA sequences using NCBI database. Freeze-dried bacterial strains were activated by culturing on a beef extract peptone medium (beef extract 3 g, peptone 10 g, sodium chloride 5 g, agar 20 g, distilled water 1 L, pH 7.2–7.4) in the dark at 30 °C for 48 h. Bacteria were amplified via culturing them in a liquid beef medium as described above except without containing an agar. Bacterial culturing was carried out under shaking at 180 rpm at 30 °C for 48 h.

*Rhizoctonia solani* was purchased from Agricultural Culture Collection of China (Strain number 36246). *R. solani* was activated on the potato dextrose agar (PDA) medium (potato 200 g, glucose 20 g, agar 20 g, water 1 L), and cultured in the dark at 30 °C for 48 h. The activated strain was inoculated into the potato dextrose broth (PDB) medium (the same formula as PDA, without agar). The *R. solani* fermentation broth was obtained after shaking cultures at 180 rpm and 30 °C for 48 h.
2.2. Inhibition of R. Solani by VOCs

The bacterial fermentation broth (100 µL) of the tested strains was spread by a sterile coating rod on the Petri dish containing beef extract peptone medium, and cultured at 30 °C for 24 h. R. solani fermentation broth (10 µL) was dropped in the center of a Petri dish containing PDA medium and placed on top of a Petri dish of PDA medium inoculated with R. solani. At the same time, a Petri dish without inoculation bacteria was set as the control. The two bottom dishes were sealed and cultured at 30 °C in the dark. The colony diameter of R. solani was recorded before the control R. solani reached the edge of the Petri dish. The treatment was repeated three times to assess the inhibitory effects of VOCs. The inhibition rate was measured as follows: Inhibition rate = [(colony diameter of control R. solani − colony diameter of treated R. solani)/colony diameter of control R. solani] × 100%.

2.3. Collection of VOCs

An aliquot of 5 mL of beef extract peptone culture medium and 0.1 g agar were added to a 20 mL headspace bottle, then sealed with kraft paper, and sterilized at 121 °C for 30 min. After sterilization, the bottles were tilted 30 °C to cool and solidify the medium. Then 100 µL of the prepared bacterial fermentation broth was added to each bottle, shaken to distribute the bacterial fermentation broth evenly on the agar slant, covered with kraft paper, and incubated at 30 °C in the dark. After 48 h, the kraft paper was replaced by a Polytetrafluoroethylene (PTFE) septum seal with a hollow screw cap. A headspace bottle containing the medium, but no bacteria, was used as the control. All treatments were repeated four times.

After culturing for 5 days, VOCs from the headspace of each vial were collected by a 50/30 µm DVB/CAR/PDMS extraction fiber (purchased from Supelco, Bellefonte, PA, USA). The extraction fiber was aged according to the manufacturer’s instructions before use and was equipped with a solid-phase microextraction (SPME) handle (Supelco, Bellefonte, PA, USA). The extraction was carried out at 30 °C for 12 h.

2.4. Analysis of VOCs

VOCs were analyzed by a 7890A-5975C GC-MS System (Agilent Technologies, Palo Alto, CA, USA) immediately after extraction. The chromatographic conditions were injection port temperature 250 °C, injection time 2.7 min, split-less mode, carrier gas 99.999% high purity helium, and column flow rate 1 mL min⁻¹. The temperature program of a column oven was initial temperature 50 °C for 2 min, an increase to 180 °C at 8 °C min⁻¹, and then further to 240 °C at 10 °C min⁻¹, this was then held at 240 °C for 6 min. The mass spectrometer (MS) conditions were electron ionization (EI) mode, 70 eV, ion source temperature 230 °C, quadrupole temperature 150 °C, transmission line temperature 250 °C, full scan mode, and the scanning range 35–450 amu. The MS of VOCs detected were identified by comparison with the NIST/EPA/NIH database.

2.5. In Vitro Verification Using Pure Compounds

The PDA medium dish edge fragment was placed in a sterilized 200 µL container. R. solani fermentation broth (10 µL) was added to the center of the medium. Pure chemical compounds (50, 100, and 200 µL) benzoic acid ethyl ester (Sangon Biotech, Shanghai, China), 3-methyl-butanoic acid (Adamas, Basel, Switzerland), 3-methyl-1-butanol (Sigma-Aldrich, St. Louis, MO, USA), 6-methyl-5-hepten-2-one (Sigma-Aldrich, St. Louis, MO, USA), 2-ethyl-1-hexanol (Sigma-Aldrich, St. Louis, MO, USA), dimethyl disulfide (Macklin, Shanghai, China), and dimethyl trisulfide (Macklin, Shanghai, China) were added to the container separately, with sterile water added if the volume was less than 200 µL (the concentrations were 25%, 50%, and 100%, 25%: 50 µL pure compound and 150 µL sterile water/plate, 50%: 100 µL pure compound and 100 µL sterile water/plate, 100%: 200 µL of pure compound/plate). Their concentrations were greater than 98%. The same amount of sterile water (200 µL) was added to the control container, and was cultivated in the dark at 32 °C to observe growth. All treatments had four replicates.
2.6. Evaluation of the Inhibitory Activity of Pure Compounds on the Lesion Development by *R. solani* on Detached Rice Leaves

*Rhizoctonia solani* was inoculated on PDA medium and pre-cultured for 48 h. The rice leaves were cut into 5-cm-long fragments and then placed on the PDA medium. Different compounds (200 µL) were added to the 200 µL container containing the plate edge fragment. The concentrations were 0%, 12.5%, 25%, and 50% (0%: 200 µL sterile water/plate, 12.5%: 25 µL pure compound and 175 µL sterile water/plate; 25% and 50% were same as above), respectively. Each treatment had five replicates and observations were made every 24 h.

2.7. Statistical Analyses

Microsoft Excel 2010 (Microsoft Corporation, Redmond, WA, USA) and SPSS for Windows version 21 (SPSS Institute Inc., Armonk, NY, USA) were used for statistical analysis. The random forest variable importance plot and Venn diagram were made in R 4.0.2, and the packages used were “randomForest” and “UpSetR”.

3. Results

3.1. Inhibition of VOCs on *R. solani*

The tested strains were *Pseudomonas* sp. (No. 3), *Enterobacter* sp. (No. 26), *Enterobacter* sp. (No. 34), *Pseudomonas* sp. (No. 35), *Ralstonia* sp. (No. 50), *Bacillus* sp. (No. 62), *Arthrobacter* sp. (No. 146), *Brevibacillus* sp. (No. 2–18), and *Paenisporosarcina* sp. (No. 2–60), respectively. They all inhibited the growth of *R. solani* by releasing VOCs, with the inhibition ranging from 7.84% to 100%. The strain No. 50 of *Ralstonia* sp. had the strongest inhibitory effect (100%). The weakest inhibitory effect (7.84%) was recorded for the strain No. 26 of *Enterobacter* sp. The inhibitory effects of the strains No. 3, No. 26, No. 34, No. 35, No. 35, No. 50, No. 62, No. 146, No. 2–18, and No. 2–60 were 36.08%, 48.63%, 39.22%, 38.04%, 20.39%, 29.80%, and 18.43%, respectively (Figure 1).
Figure 1. (A): Inhibition rate of different bacteria against *Rhizoctonia solani*. The inhibition rate was measured with the following formula: Inhibition rate = (The colony diameter of control *R. solani* − The colony diameter of treated *R. solani*)/The colony diameter of control *R. solani* × 100%. (B): Pictures of *R. solani* treated with volatile organic compounds (VOCs) from different bacteria. ck: *R. solani* treated with uninoculated beef extract peptone medium.

3.2. Composition of VOCs

A total of 89 compounds were detected across all tested strains, but different strains produced different types of VOCs. Some VOCs were unique to a specific strain, but others were produced by several strains. The number and abundance of VOCs produced by each strain were also different. The number of VOCs released by an individual strain ranged from 7 to 33. The number of VOCs released by strains No. 3, No. 26, No. 34, No. 35, No. 50, No. 62, No. 146, No. 2–18, and No. 2–60 were 7, 17, 21, 30, 10, 29, 10, 18, and 33, respectively (Table 1).

Table 1. Composition of VOCs produced by different bacterium and their peak area on mass spectrum.

| VOCs | 3   | 26  | 34  | 35  | 50  | 62  | 146 | 2–18 | 2–60 |
|------|-----|-----|-----|-----|-----|-----|-----|------|------|
|      | (× 10^8 Ab × s) |     |     |     |     |     |     |      |      |
| (E)-1-Methyl-2-(prop-1-en-1-yl)disulfane | 0.14 |     |     |     |     |     |     |      |      |
| beta-Phenylethyl butyrate                | 0.15 |     |     |     |     |     |     |      |      |
| 11-Dodecen-2-one                         |     | 1.93|     |     |     |     |     | 0.18 |      |
| 2-methyl-1-Butanol                        |     | 3.12| 4.76|     | 0.11| 0.51|     |      | 0.18 |
| 3-methyl-1-Butanol                        |     | 0.17| 0.12|     |     |     |     |      |      |
| 1-Butanol, 3-methyl-acetate               |     |     |     |     |     |     |     |      |      |
| 1-Octadecene                              |     |     |     |     |     |     |     | 0.22 |      |
| 1-Octanol                                 |     |     |     |     |     |     |     | 0.47 |      |
| 7-methyl-1-Octene                         |     |     |     |     |     |     |     | 0.30 |      |
| 1-Heptadecene                             |     |     |     |     |     |     |     |      | 0.26 |
| 2-Chloropropionic acid, hexadecyl ester   |     | 0.93|     |     |     |     |     |      |      |
| 5-heptyldihydro-2(3H)-Furanone            |     | 0.13|     |     |     |     |     | 1.23 |      |
| 2,4,6-Cycloheptatrien-1-one               |     |     |     |     |     |     |     |      |      |
| 2,4-Dithiapentane                         |     | 0.43|     |     |     |     |     |      | 0.26 |
| 6-methyl-2,4-Heptanedione                 |     |     |     |     |     |     |     |      |      |
| 3,7,11-trimethyl-2,6,10-Dodecatrrien-1-ol | 1.82| 0.12| 0.12| 0.55| 0.63| 0.49| 0.30|      |      |
| 2-Decanone                                |     |     |     |     |     |     |     |      |      |
| 2-Dodecanone                              |     |     |     |     |     |     |     |      |      |
| 2-ethyl-1-Hexanol                         |     | 0.20| 0.18| 0.68| 0.14| 0.30| 0.68| 0.97 |      |
| 2-Heptanone                               |     | 0.18| 0.25| 1.99|     |     |     |      |      |
| 3-methyl-2-Heptanone                      |     | 0.24|     |     |     |     |     |      |      |
| 5-methyl-2-Heptanone                      |     | 3.84|     |     |     |     |     |      |      |
| 6-methyl-2-Heptanone                      |     | 2.75|     |     |     |     |     |      |      |
| 2-Hexadecanone                            |     |     |     |     |     |     |     |      |      |
| 2-Hexane                                  |     | 0.12| 0.41|     |     |     |     | 0.18 |      |
| 3,4-dimethyl-2-Hexanone                   |     |     |     |     |     |     |     |      | 0.13 |
| 5-methyl-2-Hexanone                       |     |     |     |     |     |     |     |      | 0.92 |
|
| VOCs | 3   | 26  | 34  | 35  | 50  | 62  | 146 | 2–18 | 2–60 |
|------|-----|-----|-----|-----|-----|-----|-----|------|------|
|      | ($\times 10^8 \text{ Ab } \times \text{s}$) |
| 2-Nonanone | 0.48 | 0.40 | 0.71 | 0.87 | 0.40 |
| 2-Octanone |       |       |       | 0.26 | 0.45 |
| 2-Pentadecanone | 0.14 |       |       |       |       |
| 2-Pentanone | 0.13 | 0.55 |       |       |       |
| 3-methyl-2-Pentanone |       |       |       |       |       |
| 2-Tetradecanone | 0.19 |       |       | 0.89 | 1.88 |
| 2-Tridecanone |       |       |       | 0.26 |       |
| 2-Undecanone | 0.62 | 0.55 |       | 1.18 | 0.63 |
| 3-Dodecanone |       |       |       |       | 0.20 |
| 7-phenyl-3-Heptene | 0.19 |       |       |       |       |
| 5-methyl-3-Hexanone | 3.58 |       | 0.90 |       | 0.57 |
| 3-Pentadecanone |       |       |       | 0.65 |       |
| 3-Pentanone |       |       |       | 0.44 |       |
| 3-Tridecanone | 0.52 |       |       | 0.12 |       |
| 5-methyl-4-Hexen-3-one |       |       |       | 0.90 |       |
| 6,10,14-trimethyl-5,9,13-Pentadecatrien-2-one | 3.20 |       |       | 0.16 | 0.15 |
| 6-methyl-5-Hepten-2-one |       |       |       |       | 0.11 |
| 6-tert-Butyl-2,4-dimethylphenol | 0.52 | 0.95 |       |       |       |
| 7-Methyloctane-2,4-dione, enol form |       |       |       |       | 0.24 |
| Acetic acid, 2-phenylethyl ester | 0.31 |       |       |       |       |
| Acetic acid, chloro-, hexadecyl ester | 0.29 |       |       |       |       |
| Acetic acid, non-3-enyl ester, cis- (2-methoxyethyl)-Benzene | 1.43 |       |       |       |       |
| (methoxymethyl)-Benzene | 1.35 |       |       |       |       |
| Benzeneacetic acid, ethyl ester Benzoic acid, ethyl ester |       | 1.74 |       |       |       |
| Benzy l alcohol |       | 1.59 | 0.44 |       |       |
| Benzyl methyl ketone | 0.14 | 0.79 | 0.13 | 0.52 | 0.11 |
| 1-methoxy-3-methyl-Butane | 9.96 |       |       |       |       |
| 2-methyl-2-(methylthio)-Butane |       |       | 0.63 |       |       |
| Butanethioic acid, 5-methyl ester Butanoic acid, 1-ethenylhexyl ester |       | 0.04 | 0.25 | 0.74 |
| 3-methyl-Butanoic acid | 0.45 | 1.87 | 0.16 |       |       |
| Butanoic acid, 3-methyl-, ethyl ester cis-Bicyclo [3.3.0]oct-2-ene |       | 0.12 | 0.42 | 0.83 |
| Cycloheptene | 0.43 |       |       |       |       |
| 1-methyl-Cyclohexene |       | 0.24 |       |       |       |
| 3-ethyl-Cyclopentene |       |       |       |       | 0.17 |
| Dicyclopentadiene |       | 0.16 | 0.24 | 0.14 | 0.29 | 0.18 |
| Dimethyl trisulfide | 0.32 | 0.20 | 0.19 | 1.98 | 1.67 | 1.96 | 0.87 |
| Disulfide dimethyl | 18.7 | 1.18 | 1.33 | 2.23 | 16.1 | 3.13 | 9.98 | 3.21 |
| Dodecanoic acid, ethyl ester 1-(2-aminophenyl)-Ethanone |       | 0.13 | 0.16 |       |       |
| Ethyl 13-methyl-tetradecanoate |       |       |       |       | 0.23 |
| Ethyl 3-(methylthio)-(E)-2-propenoate |       |       |       | 0.23 |       |
| Ethyl 3-(methylthio)-(Z)-2-propenoate |       |       |       | 0.26 |       |
| Ethyl tridecanoate |       |       |       |       |       |
| Methyl Isobutyl Ketone | 1.29 | 0.42 |       |       |       |
| Methyl isovalerate | 0.58 |       |       |       |       |
| Methyl thiolacetate decahydro-Naphthalene | 2.62 | 0.44 | 0.84 |       |       |
| Octadecan 11 |       | 0.12 |       |       |       |
| Phenylethyl Alcohol | 1.69 | 1.63 | 0.15 | 0.57 |       |
| Propanoic acid, 2-phenylethyl ester Pyrazine, 2-ethyl-5-methyl-trimethyl-Pyrazine |       | 0.15 | 0.24 | 0.50 |
| Pyrrole |       |       |       |       | 0.41 |
| S-Methyl 3-methylbutanethioate TATP | 0.17 | 3.70 | 0.34 | 0.15 | 3.72 |
| Tetradecanoic acid, ethyl ester |       |       | 0.54 |       |       |
| 2-methoxy-5-methyl-Thiophene |       | 0.35 |       |       |       |
| Undecanoic acid, ethyl ester |       |       |       |       | 0.24 |

There were 15 compounds unique to strain No. 2–60, six unique to No. 35, seven to No. 62, six to No. 34, five to No. 2–18, eight to No. 26, six to No. 50, two to No. 146, and
three unique to No. 3. One compound was released by eight strains, one compound was released by seven strains, five compounds were released by six strains, three compounds were released by five strains, three compounds were released by four strains, seven types of compounds were released by three strains, and 12 compounds were released by two strains (Figure 2).

Figure 2. The venn diagram showing the amount of different and same VOCs released by different strains. The left side shows the total number of VOCs released by each strain, and the above shows the number of VOCs unique to each strain or shared with other strains.

3.3. Classification and Abundance of VOCs Emitted from Different Bacterial Strains

The VOCs released by all tested strains were mainly esters, olefins, alkanes, ketones, acids, thiophenes, aldehydes, naphthalenes, sulfides, phenols, alcohols, pyrazines, pyrroles, and benzenes. All strains released alcohols, but strains No. 34 and 62 released the most amounts. Strains No. 3, 35, 50, 146, and 2–60 released the largest amounts of sulfides, strain No. 26 released the largest amounts of alkanes, while strain No. 2–18 released the largest amounts of ketones (Figure 3).

Figure 3. Types and normalized peak area of VOCs emitted from different bacteria.
3.4. The Relative Importance of Different VOCs

The relative importance of different VOCs was estimated by constructing a random forest model. It was found that sulfides, olefins, alkyls, ketones, and esters were more important than the other types of VOCs. The ranking was based on two criteria: Mean Square Error and Node Purity. The proportion of variance explained by this model was 94.23% (Figure 4A,B). In addition, a random forest model of the importance of different individual VOCs was established. This model explained 95.24% of the variance. Using this model, the top 10 most important VOCs were ranked. These compounds with decreasing importance were 2-methyl-2-methylthio-butane, benzoic acid ethyl ester, 6,10,14-trimethyl-5,9,13-pentadecatrien-2-one, 3-methyl-butanolic acid, and benzyl methyl ketone (Figure 4C,D).

Figure 4. Random forests variable importance plot of different types VOCs ((A): % IncMSE, (B): IncNodePurity)) and random forests variable importance plot of different individual VOCs ((C): % IncMSE, (D): IncNodePurity). % IncMSE: Relative importance.

3.5. The Inhibition of R. solani by Pure Compounds

Benzolic acid ethyl ester, 3-methyl-butanolic acid, 2-ethyl-1-hexanol, and dimethyl trisulfide were lethal to R. solani, whereas 3-methyl-1-butanol and 6-methyl-5-hepten-2-one significantly inhibited the growth of R. solani at 25% concentration. At 50% and 100% concentrations, all compounds except dimethyl disulfide were lethal to R. solani. However, dimethyl disulfide also significantly inhibited the growth of R. solani (Figure 5).

Benzolic acid ethyl ester had the strongest suppression on R. solani infection, with nearly complete inhibition under all tested concentrations greater than 12.5%. 2-Ethyl-1-hexanol inhibited the infection of detached leaves by R. solani at the concentration of 12.5%. The effect of benzolic acid ethyl ester became increasingly obvious as the concentration of benzolic acid ethyl ester increased. By contrast, 3-methyl-butanolic acid showed only a slightly inhibitory effect on R. solani at all tested concentrations, but inhibition did not reach very high levels. (Figure 6).
Figure 5. The inhibitory effect of pure substances on R. solani. 25% concentration was 50 µL pure substances and 150 µL sterile water/plate (A), 50% concentration was 100 µL pure substances and 100 µL sterile water/plate (B), and 100% concentration was 200 µL of pure chemical substance/plate (C). CK was 200 µL sterile water/plate. (D): Chemical structures of purified compounds.

Figure 6. The inhibitory activities of pure compounds on the lesion development by R. solani on detached rice leaves.
4. Discussion

In this study we found that nine bacterial strains isolated from the rice rhizosphere inhibited the growth of *R. solani* without direct contact (a double Petri dish assay), indicating that these bacteria released VOCs that were inhibitory to the growth of *R. solani* (Figure 1). Although there are many related reports in the literature on bacteria that could inhibit the growth of pathogens by releasing VOCs [21,22], no report can be found on the VOCs released by rice rhizosphere bacteria that can inhibit the growth of *R. solani*, specifically. Here we discovered nine bacterial strains that can release VOCs, which can then elicit inhibitory effects on *R. solani*. Based on sequencing and comparison in the NCBI database, we found these bacterial strains belong to *Pseudomonas* sp. (No. 3), *Enterobacter* sp. (No. 26), *Enterobacter* sp. (No. 34), *Pseudomonas* sp. (No. 35), *Ralstonia* sp. (No. 50), *Bacillus* sp. (No. 62), *Arthrobacter* sp. (No. 146), *Brevibacillus* sp. (No. 2–18), and *Paenisporosarcina* sp. (No. 2–60), respectively. As expected, different strains exhibited different inhibitory effects on *R. solani*, ranging from 7.84% to 100%. Surprisingly, strain No. 50 of *Ralstonia* sp. showed 100% inhibition of *R. solani* growth (Figure 1). In comparison, Mookherjee et al. found that the endophyte *Geotrichum candidum* utilized different carbon sources to varying extents [23], and the antifungal activity of its VOCs against phytopathogen *R. solani* changed with each carbon source, with the maximum mean growth inhibition of about 62% observed with glucose. Elkahoui et al. (2015) found 25 VOCs producing strains inhibited (by 27–50%) the growth of *R. solani* [24]. However, no previous study has found a strain that would kill *R. solani*. Hence, the present study is the first one to document complete (100%) inhibition of *R. solani* growth by bacterial VOCs produced by *Ralstonia* sp. No. 50.

The profiles of VOCs released by the nine bacterial strains were also analyzed in our study. We found that some VOCs were commonly released by different bacterial strains, while others were unique to each strain (Table 1, Figure 2). In addition to the differences of VOC profiles, the abundance of different VOCs was also different. These findings are consistent with those reported previously by Li et al. [25], who tested VOCs released by eight *Bacillus* strains, which released a variety of VOCs in different amounts. Hernández-León et al. also found that four *Pseudomonas fluorescens* UM16, UM240, UM256, and UM270 produced different VOCs in different concentrations [26]. The different bacterial strains characterized in the different studies are the likely reason for the differences in the type and content of VOCs observed.

In the random forest variable importance plot of different VOCs, benzoic acid ethyl ester and 3-methyl-butanoic acid contributed more to *R. solani* inhibition than 2-ethyl-1-hexanol, 3-methyl-1-butanol, and 6-methyl-5-hepten-2-one. Benzoic acid ethyl ester and 3-methyl-butanoic acid were found lethal to *R. solani* at 25% concentration. The VOCs 3-methyl-1-butanol, and 6-methyl-5-hepten-2-one were able to significantly inhibit the growth of *R. solani* at 25% concentration, but their contribution was less than that of benzoic acid ethyl ester and 3-methyl-butanoic acid. Therefore, it may be reasonable to speculate that 2-methyl-2-(methylthio)-butane and 6,10,14-trimethyl-5,9,13-pentadecatrien-2-one can also totally inhibit *R. solani* under different conditions. These two VOCs have not been reported hitherto to have antifungal activity previously and their antifungal activity remains to be investigated more extensively and vigorously.

2-Ethyl-1-hexanol was also lethal to *R. solani* at 25% concentration. Similarly, benzoic acid ethyl ester, 3-methyl-butanoic acid, 2-ethyl-1-hexanol, 3-methyl-1-butanol, and 6-methyl-5-hepten-2-one, which had similar contributions in the random forest model, all showed lethality or significant inhibition at 25% concentration, and were definitely lethal at 50% and 100% concentrations (Figures 4 and 5). In the literature, some other VOCs, such as 2-phenylethanol, isopentyl acetate, naphthalene, methyl 2-methylpentanoate, 1,3,5-trichloro-2-methoxy benzene, 1-methyl-4-(1-methylethenyl)-cyclohexene, and 4-flavanone (4H-1-benzopyran-4-one, 2,3-dihydro-2-phenyl), were also reported to inhibit the growth of *R. solani* [19,23,27]. These VOCs were not detected in the present study, possibly because we isolated bacteria from the rhizosphere of rice, whereas the previous studies used other bacterial sources.
Several studies have described antifungal activity of bacterial VOCs, but few have identified a single VOC or a blend of VOCs responsible for the antifungal activity [28]. Fernando et al. used commercial VOCs for in vitro verification of their antifungal activity [29], and only a few of them were found effective. This approach is time-consuming, laborious, and costly [30,31]. In our study a random forest model was constructed to explore the relationship between VOCs and antifungal activity for the first time. This was an easy and effective way to compare the contribution of different VOCs to antifungal activity and identify the most effective VOCs. Tests of pure commercial compound showed that estimation based on the random forest model was reliable. This method saved time, labor, and money.

It is very interesting to note that some of the most potent VOCs were easily synthesized and are commercially available, such as benzoic acid ethyl ester, 3-methyl-butanoic acid, 2-ethyl-1-hexanol, 3-methyl-1-butanol, 6-methyl-5-hepten-2-one. In addition, we found that dimethyl trisulfide was lethal to *R. solani* at 25% concentration. Benzoic acid ethyl ester, 2-ethyl-1-hexanol, and 3-methyl-butanoic acid were also lethal to *R. solani* under certain conditions. These three compounds could actually inhibit *R. solani* infection on detached leaves (Figures 5 and 6). However, the previous studies used pure compounds for verification, and no study has ever used smaller concentrations of compounds to verify the antifungal activity. Hence, the VOCs found in this study showed high antifungal activity. However, we assume that the VOCs produced by the rhizosphere bacteria when grown on a beef peptone medium would also be produced when they colonize roots of the rice plant. That may or may not be the case, as nutrition obtained from a complex medium may result in different products vs. those when grown in a natural system. Nonetheless, our findings are interesting. We also identified volatiles produced by the different bacterial strains as well as evaluated the impact those volatiles have on disease suppression. Again, this was done under “laboratory” conditions that do not necessarily reflect the real world, but there is enough information presented for us or others to pursue this line of investigation. For example, the potential interaction of production of the VOCs in a complex environment interacting with many potential feedback mechanisms. Therefore, the system presented here is not ready to release as a commercially ready biocontrol, but the data presented here are intriguing and sufficiently represent a starting point to have a decent system for further study.

5. Conclusions

*Pseudomonas* sp. No. 3, *Enterobacter* sp. No. 26, *Enterobacter* sp. No. 34, *Pseudomonas* sp. No. 35, *Ralstonia* sp. No. 50, *Bacillus* sp. No. 62, *Arthrobacter* sp. No. 146, *Brevibacillus* sp. No. 2–18, and *Paenisporosarcina* sp. No. 2–60 were nine strains of bacteria found for the first time to inhibit *R. solani* growth via VOCs. The inhibitory effect ranged from 7.84% to 100%, with *Ralstonia* sp. No. 50 inhibiting the growth of *R. solani* completely (100%). Benzoic acid ethyl ester, 3-methyl-butanoic acid, 2-ethyl-1-hexanol, 3-methyl-1-butanol, and 6-methyl-5-hepten-2-one were VOCs released by bacteria in the rice rhizosphere. These five VOCs were found to be toxic to *R. solani* for the first time. Among them, benzoic acid ethyl ester, 3-methyl-butanoic acid, and 2-ethyl-1-hexanol were lethal to *R. solani*. Overall, novel functions of bacterial strains from the rice rhizosphere and their VOCs have the potential to be used as resources for biological control of rice sheath blight in the future.

**Author Contributions**: Conceptualization, F.F., A.S. and E.W.; methodology, E.W., J.B., S.W. and J.Z.; formal analysis, E.W., Z.S., X.L. (Xu Li) and W.D.; writing—original draft preparation, E.W.; writing—review and editing, F.F., A.S., M.C., and X.L. (Xiongduo Liu). All authors have read and agreed to the published version of the manuscript.

**Funding**: The research was funded by the National Key Research and Development Program of China (2019YFD1002004) and Fundamental Research Funds for Central Non-profit Scientific Institution (No. 1610132019011, 1610132020012).
Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: The data presented in this study are available on request from the corresponding author.

Acknowledgments: We thank the reviewers for their valuable comments and recommendations. We also thank Timothy George for polishing the language.

Conflicts of Interest: The authors declare no conflict of interest.

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