**INTRODUCTION**

Microbial infection and its treatment is a major concern today. There is increasing drug resistance to the existing antibacterial agents paved to develop the novel treatment strategies. Currently, several new approaches are adopted to manage the toxicity; multidrug resistance arises due to bacterial infections such as antibiotic-resistant, antimicrobial peptides (AMPs), vaccination, phage therapy, bacteriocins, killing factors, non-antibiotic antibacterial drugs, and quorum sensing inhibitors [1,2]. Recent studies indicate that AMPs could be a better alternative to combat the multidrug resistance infections [2]. AMPs are oligopeptides containing a varying number of amino acids from 5 to more than 100 amino acids. AMPs have a broad spectrum of activity. They target virus, fungi, bacteria, and parasite.

The rapid increase in antibiotic resistance and their short spectrum of activity lead to the emergence of AMPs. These AMPs have proved to be effective against the “superbugs”. The AMPs have a broad spectrum of activity, rapid action, and show synergic activity with other old antibiotics, AMPs, and lysozymes. AMPs block the release of cytokines by bacterial products in tissue culture and human blood and the onset of sepsis is also blocked by them in the mouse model endotoxemia [3].

The AMPs can be derived from both plants as well as animal sources. Dubos in 1939, extracted an antimicrobial agent from a soil *Bacillus* strain. This extract was witnessed to protect mice from *Pneumococci* infection [4]. In the impending years, Hotchkiss along with dubos fractionated that extract and found an AMP which was named as gramicidin [5]. Besides toxicity on the intra peritoneal application, gramicidin was proved to be effective for topical treatment of wounds and ulcers [5,6]. Moreover, in 1941 tyrocidine was discovered which was found to be effective against Gram-negative as well as Gram-positive bacteria [7]. However, tyrocidine was found to cause hemolytic activity [8]. In the same year, purothionin was extracted from plant *Triticum aestivum* and found to be effective against fungi and some pathogenic bacteria [9].

Apart from these plant-derived AMPs, animal-based AMPs were also discovered. The first animal derived AMP was defensin, which was extracted from rabbit leukocytes in 1956 [10]. In the impending years, bombinin from epitheliaβ and lactokerrin from cow milk [11] were described. In the same year, it was proved that human leukocytes contain AMPs in their lysosomes [12].

MP196 is a hexapeptide (*RWRWRW-NH*) which represent the small peptide with hydrophobic and positively charged amino acids as key pharmacophore depicted in (Fig. 1) [13,14]. MP196 is showed good activity against all type of bacteria but more effective against Gram-positive bacteria including methicillin-resistant *Staphylococcus aureus* strains, also it found less toxic with low hemolytic activity [15]. Hence, MP196 could serve as the potential lead for further improvement in AMPs.

**METHODS**

Molecular modeling is a very much investigated technique for recognizing the potent compound without putting excessively exertion and investment in research [16-20]. ADT 1.5.6. software [21] is used by us to investigate the activity in terms of binding affinity (Kcal/mol), and there after the outcomes are compared in binding affinity score for best-docked conformation. The structures of various AMPs were drawn by the help of ChemBiodraw ultra and further converted to the 3D structure using ChemBiodraw 3D. All the designed structures were optimized by energy minimization using MM2 method [22] and converted to readable format at the ADT interface. To identify the potential AMPs, a protein 3vma was selected which was downloaded from protein data bank [23]. The outcomes of results were analyzed by AutoDock Vina (ADT) result which reveals close contact, hydrogen bond, hydrophilic, and hydrophobic interactions.
RESULTS AND DISCUSSION

We have designed the novel AMPs (KP_01R to KP_04S) on the basis of MP196 AMPs to find out the better lead compound. For the design of novel molecules, various structural changes were made on the MP196, i.e., the amino acids arginine (R) and tryptophan (W) at both terminals are replaced by leucine (L), phenylalanine (F) as well as reversing the stereochemistry of these amino acids to transform them as better ligands by keeping an amidation at C-terminal as depicted in (Fig. 2) [24].

Replacement of terminal amino acids leads to a change in the net charge as well as lipophilicity of the designed AMPs by using non-polar amino acids like leucine and phenylalanine. The designed molecules

![Fig. 1: MP196 an antimicrobial peptide](image1)

![Fig. 2: Design of novel antimicrobial peptides-based MP196](image2)
were analyzed to identify the most potent AMP by molecular modeling software, so selected 3vma protein of *Escherichia coli* was prepared by recognizing the binding site.

First, the validation of protein was done by extraction of ligand (Fig. 3) and after extraction of ligand from the protein; it was prepared for the docking study by adding the polar hydrogen, detecting root, and converting it to pdbqt extension file. For docking study, after extraction of ligand, 3vma protein is prepared by removing water molecules, repairing missing atoms, adding polar hydrogen only, and subsequently adding the Kollman charges. Further, the grid box was generated keeping the ligand as a center (Fig. 4).

From grid output file the configuration file “conf.txt” was prepared and command prompt was used for ADT molecular docking by giving command “program files\the scripps research institute\vina\vina.exe - config conf.txt - log log.txt” It generated the output file with the docking score or binding affinity (Kcal/mol), similarly, all the designed molecules were studied, and their binding affinities are showed in Table 1.

The results revealed that all the designed AMPs showed better binding affinity ranging from −8.1 to −9.0 Kcal/mol in comparison to MP196 AMP (−7.5 Kcal/mol, Table 1). Among them, KP_03R found to be best with −9.0 Kcal/mol (Table 1) at 3vma, penicillin-binding protein of *E. coli*. In-depth analysis at the binding site of 3vma (Fig. 5), it is observed that KP_03R has two hydrogen bonds at the distance of 2.246 and 2.043 with the energy of −2.9 and −6.1 Kcal/mol, respectively (Fig. 6). The hydrogen bonds involved between the protein and KP_03R are hydrogen of Arg 286 and carbonyl of Asn 275 residue. Whereas, its antipode or other *S* stereoisomer KP-03S showed a less binding affinity (−8.5 Kcal/mol, Table 1). Similarly, the stereochemistry affects the binding affinity of other designed AMPs suggesting that “*R*” stereoisomer would be more effective than “*S*” stereoisomers (Table 1).

Further, we examined the various physical properties of the peptide by using the online peptide data bank using the basic amino acid sequence of KP_03R AMP. The amino acid residues sequence of this peptide was submitted to the AMP database (APD2) a predictive tool based on APD2 v2.34 (APD2; http://aps.unmc.edu/AP/main.php) [25,26]. The result

| AMPs   | Binding affinity (Kcal/mol) |
|--------|-----------------------------|
| MP196  | −7.5                        |
| KP_01R | −8.8                        |
| KP_01S | −8.5                        |
| KP_02R | −8.9                        |
| KP_02S | −8.3                        |
| KP_03R | −9.0                        |
| KP_03S | −8.5                        |
| KP_04R | −8.2                        |
| KP_04S | −8.1                        |

**Table 1: Designed antimicrobial peptides based on MP196 and their binding energies**

**Table 2: Calculated peptide parameters and the APD-based prediction of properties of KP_03R (FWRWRW-NH2)**

| Parameters                          | Values    |
|-------------------------------------|-----------|
| Amino acids residues                | F-W-R-W-R-W-NH2 |
| Percentage of each amino acid       | Phe (F) ratio = 16% |
|                                     | Trp (W) ratio = 50% |
|                                     | Arg (R) ratio = 33% |
| Hydrophobic ratio                   | 66%       |
| Total net charge                    | +2        |
| GRAVY: Grand average hydropathy value of the peptide | −1.48 |

**Fig. 3: Extracted ligand from 3vma protein**

**Fig. 4: Set grid box on 3vma protein as a macromolecule**

**Fig. 5: Close contacts of KP_03R (atom color in ball and stick model) with neighboring amino acid residues of 3vma protein (pink color in ribbon structure)**

**Fig. 6: Visualization of active binding sites of 3vam protein and interaction with KP_03R (a) showed the distance of hydrogen bond and (b) showed hydrogen bond energies**
in Table 2 showed optimum hydrophobicity with 66%, total net charge +2 and grand average hydrophathy value of the peptide −1.48 for KP_03R suggesting that it could be a potent AMP.

CONCLUSION

Antimicrobial drug resistance is a major concern in current and AMP could serve a better alternate to overcome it. MP196 a short chain consisting six amino acids AMP was selected for further improvement with modifying the terminal amino acid as well as their stereochemistry. The eight different AMPs were designed, and among them, KP_03R (FWRRWR-NH₂) showed the maximum activity against 3vma protein (A penicillin-binding protein from E. coli) with the binding affinity of −9.0 Kcal/mol. Moreover, the comparison was also made between R and S stereoisomers of the peptides. The S-stereoisomer KP_03S proved to be less active than R-stereoisomer KP_03R against 3vma receptor protein. Further investigations on these novel AMPs will provide the promising tool for new drug development to treat microbial infection and provide an alternative in case of drug resistance.

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