Dodecamer d-AGATCTAGATCT and a Homologous Hairpin form Triplex in the Presence of Peptide REWER

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Introduction

The considerable interest in DNA triple helix arose due to its role in human diseases [1,2] and its potential use in targeted gene therapy [3–5]. Triplex-forming oligonucleotides (TFO) can be designed that recognizes duplex DNA target in a sequence-specific manner. Studies have shown that intramolecular triplexes in which a single strand of DNA folds back into a triple-helical structure can form in vivo [6]. Purine stretches in the double helix are essential for triplex helix because pyrimidine do not have enough donor and acceptor groups to form both Watson-Crick and Hoogsteen hydrogen bonds simultaneously. Several groups used designed deoxyribonucleotide to demonstrate that intermolecular triplexes where the third strand occupies the major groove of the duplex could form at sites that are purine-rich in one strand. In these triplexes the third strand can be either parallel or antiparallel. In Pu-PuPy triplex type purine-rich third strand is hydrogen bonded to underlying purine strand of duplex in antiparallel orientation whereas pyrimidine-rich third strand in Py-PuPy triplex forms parallel triplex with purine strand of duplex. In third type of triple-helical structure the third strand pair first with purines on one strand and then switches strand to pair with purines on the other strand [6]. Recombination triplexes that form in vivo in the presence of proteins like RecA in bacteria [7] and RAD51 in eukaryotes [8] are different from the conventional triplexes observed experimentally. Contrary to the known antiparallel triplexes the third strand here is accommodated in the major groove of DNA duplex in parallel orientation of like strands with deoxyriboses in north conformation [9,10]. Several models of major groove associated parallel triplex structure have been proposed [11,12].

In a previous study an antiparallel triplex formation between synthetic deoxyoligonucleotide hairpin RY28 molecule with a homologous dodecamer YR12 was reported [13]. It was shown that RY28 does not form triplex structure with RY12, that has the same sequence but opposite polarity to that of YR12 dodecamer [13]. Authors showed that a spontaneous triplex formation is permitted only when identical strands are oriented antiparallel but not permitted in parallel orientation as in R-form DNA. Due to chosen sequences the triplex between RY28 and RY12 would be parallel in Pu-PuPy and antiparallel in Py-PuPy. Such parallel triplex is formed only under forced strand orientation or with the assistance of a protein like RecA [14]. Our synthetic oligomers (RY12 and RY28) contain the same alternate repeats of three purines and three pyrimidines so that all the strands are homologous. Any triplex formed between two oligonucleotides must involve pairing with both the strands of Watson-Crick (W-C) duplex. We now report here the evidence for the formation of a triplex between RY12 and RY28 mediated by a designed pentapeptide, Arg-Glu-Trp-Glu-Arg (REWER).

Abstract

We have designed a dodecamer d-AGATCTAGATCT (RY12) with alternate oligopurines and oligopyrimidines tracts and its homologous 28 bp hairpin oligomer (RY28) that forms a triple helix only in the presence of a pentapeptide REWER. An intermolecular triplex is formed by the single strand invasion of the RY28 duplex by RY12 in the presence of REWER. 5'-oligopurine end of RY12 binds to oligopurine sequence of RY28 in a parallel orientation and its oligopyrimidine stretch then changes strand and adopts an antiparallel orientation with the other strand of the duplex. Evidence for the formation of the triplex come from our studies of the UV melting curves, UV mixing curves, gel retardation assay, and chemical sequencing of 1:1 mixture of dodecamer and hairpin oligonucleotides in the presence and absence of the peptide REWER. RY12 exists as a duplex that melts at 35 °C. The hairpin (RY28) melts at 68 °C. 1:1 mixture of RY12 and RY28 in the absence of REWER gives a biphasic transition curve with thermodynamic properties corresponding to those of the melting of the duplex of RY12 and the hairpin RY28. However, the melting curve of this mixture is triphasic in the presence of the REWER; the thermodynamic parameters associated with the first phase (melting of the duplex of RY12), second phase (melting of the triplex) and the third phase (melting of the hairpin) show dependence on the molar ratio of peptide to oligonucleotides. Under appropriate conditions, gel retardation assay showed a shifted band that corresponds to a possible triplex. Chemical sequencing of KMnO₄ and DEPC treated mixture of RY12, RY28 and REWER revealed the footprint of triplex.
Materials and Methods

Oligonucleotides and peptide synthesis

Synthesis and purification of RY12 (AGATCTAGATCT-3') and RY28 (AGATCTAGATCTCTAGATCTAGATCT-3') deoxyoligonucleotides were carried by solid phase β-cyanoethylphosphoramidite chemistry [15]. Deprotection and purification of the dodecamer oligonucleotides were done by standard procedure [16]. We designed peptide REWER such that the tryptophan is placed in the middle of the pentapeptide and there are two glutamic acids and two arginines flanking it making this peptide a charge-neutral molecule. The pentapeptide, REWER was synthesized by solid phase peptide synthesis using F-moc chemistry on a semi-manual peptide synthesizer (Novasyn, France). The crude peptide was purified on reverse phase column (GE Healthcare, USA). The peptide synthesis reagents were from Novabiochem (USA). All solvents and reagents were of analytical grade from Sigma Chemical Co. (USA). Stock solutions of oligonucleotides RY12 and RY28 and peptide, REWER were prepared in Buffer A (10 mM Na-acacodylate buffer (pH 5.8) with 100 mM NaCl and 10 mM MgCl₂). Concentrations of the stock solutions of REWER, RY28 and RY12 were determined using absorbption coefficient (M⁻¹ cm⁻¹) values of 5,700, 9,000, and 10,000, respectively.

Optical studies

For spectroscopy measurements samples with appropriate concentration of oligonucleotides and peptide (4.5x10⁻⁵–5.5x10⁻⁵ M) were prepared in degassed buffer. Thermal denaturation measurements were carried out in JASCO UV-Vis 560 spectrophotometer with a heating rate of 1°C per minute using a Peltier accessory (Model ETC-505). The change in absorbance of each sample was measured at 260 nm, and approximately 350 data points were collected for each denaturation curve. Prior to melting study of oligomer mixtures, solutions were heated to 90°C for 2 minutes and annealed by slowly decreasing the temperature to 0°C. We found that these annealing conditions facilitated formation of the interacting complex and resulted in a well-defined melting profile. The reversibility of the transition was checked by matching the absorbance at 260 nm and 25°C before and after heating.

For UV mixing experiments equimolar (30 μM each) mixture of RY12 and RY28 were prepared in the degassed buffer A. Each solution contained the peptide REWER and nucleotides at a molar concentration of the peptide and total DNA strand solution contained the peptide REWER and nucleotides at a ratio of [P]/[N] = 3.8. All solutions in Buffer A were heated to 90°C for two minutes, followed by slow annealing at room temperature. 5 μl of calf thymus DNA (CTD) (1 mg/ml), 100 μl of 0.5 M NH₄(OAc) and four volumes of ethanol were added to each sample to precipitate the complex overnight at -70°C. The precipitates were washed twice with 70% ethanol followed by drying. The samples were then dissolved in loading dye and immediately resolved on 15% native polyacrylamide gel (PAGE) that contained excess amount of free REWER in the gel itself. PAGE was run at 150 V in 0.25× TBE buffer for 3 hours in a cold room. The same experiment was also performed using labeled RY12 and unlabeled RY28. Since the precipitation of 12 nt DNA is difficult, radiolabeled RY12 oligomer was precipitated in the presence of 20 μl of CTD (1 mg/ml) and 200 μl of 0.5 M NH₄(OAc) and four volumes of ethanol.

Sequence of Triplex structure

Chemical modifications using potassium permanganate (KMnO₄) (Sigma Chemical Co., USA) and diethyl pyrocarbonate (DEPC) were carried out at [P]/[N] ratio of 3.8. KMnO₄ sequencing reactions were carried out on free RY28 oligomer and RY28-RY12 complex in the presence and absence of peptide REWER as per published procedure with slight modifications [19]. Approximately 300 ng of RY28 oligomer was radiolabeled with [γ²³⁵P]-ATP at the 5’ end. All samples were dissolved in the reaction Buffer A, heated to 90°C and then slowly annealed at room temperature. 5 μl of 10 mM KMnO₄ diluted freshly from the 100 mM stock at 4°C, was then added and the reaction was continued for 10 minutes at 4°C. After chemical modifications, each oligonucleotide sample was precipitated by ethanol in the presence of high salt and then subjected to pyrolytic cleavage [20], followed by loading on a 20% denaturing sequencing PAGE containing 8% urea. Bands were resolved by gel electrophoresis at 1200 V for 3 hours. For DEPC modification, 5°C end radiolabeled RY28 oligonucleotide solution in the reaction Buffer A was cooled to 4°C and then 5 μl of DEPC (Sigma-Aldrich, USA) was added and incubated at 4°C for 30 minutes. The subsequent precipitation, pyrolytic cleavage and electrophoresis on 20% denaturing PAGE with 8% urea was done as described for KMnO₄ reaction.

Results

Analysis of UV thermal transitions

Figure 1A (curve 1) shows the melting of the RY12 duplex. This process is reversible. Assuming a two-state mechanism, ΔG*(T), the Gibbs energy change at any temperature T K, associated with the melting of the duplex (double strand (d)↔ss 2 (single strand)) is determined with the help of following relation,

ΔG*(T) = −RT ln \left[ \frac{2C_0(\gamma(T)−\gamma_d(T))^2}{(\gamma(T)−\gamma_d(T))(\gamma(T)−\gamma(T))} \right] \tag{1}

where R is the gas constant; C₀ is the total strand concentration; \gamma(T) is the absorption at T K; and \gamma_d(T) and \gamma(T) are the optical properties of the duplex (pretransition region) and single strand (posttransition region) at T K, respectively. Values of ΔG*(T) in the range −1.3<ΔG*(T), kcal mol⁻¹<1.3 were plotted as a function of temperature. This plot was used to determine T_m (midpoint of melting transition of the duplex) and ΔH_m (enthalpy change on melting of the duplex at T_m) using the procedure described elsewhere [21]. These values are given in Table 1.

Figure 1A (curve 2) shows the melting transition curve of hairpin RY28. This process (native hairpin (hₙ) state↔unfolded hairpin (hₐ) state) is reversible. Assuming a two-state mechanism, ΔG*(T),
the Gibbs free energy change at any temperature T K, associated with the melting of the hairpin is determined with the help of the relation,

$$\Delta G_0(T) = -RT \ln \left[ \frac{y(T) - y_a(T)}{y_a(T) - y(T)} \right]$$  \hspace{1cm} (2)$$

where $y(T)$ is the absorption at T K, and $y_a(T)$ is the absorption of the native hairpin (pretransition) and $y_a(T)$ is the absorption of unfolded [posttransition region] forms of the hairpin. Values of $\Delta G_0(T)$ in the range $-1.3<\Delta G_0(T)<1.3$ were determined. These values of $\Delta G_0(T)$ were plotted as a function of temperature and were used to determine $T_m^{\text{h}}$ and $\Delta H_m^{\text{h}}$ as described above. All such values are given in Table 1.

Figure 1A shows the melting transition curve of 1:1 mixture of RY12 and RY28 (curve 2). This melting curve is biphasic and reversible. RY12 and RY28 do not interact and melt independently of each other. Assuming that the first phase (i.e., the process in the lower temperature range) represents the melting of the duplex (RY12) and the second phase (i.e., the process in the higher temperature range) is the unfolding of the hairpin (RY28), they were analyzed for $T_m^d$, $\Delta H_m^d$, $T_m^h$, and $\Delta H_m^h$ using equation 1 and equation 2 and the methods described above. It should be noted that the posttransition region of the first phase is the pretransition region of the second phase. Thermodynamic parameters are given in Table 1.

We also measured heat-induced transitions of the 1:1 mixture of RY12 and RY28 in the presence of peptide REWER at different mole ratios of [P] and [N], where square brackets represent the molar concentrations of the peptide [P] and the nucleotide [N], respectively. All transitions were reversible and triphasic. Figure 1B shows representative transition curves of mixtures. It is assumed that the first and the third phases represent melting of the duplex and hairpin, respectively, and they were analyzed for $(T_m^d)$ and $\Delta H_m^d$ and $(T_m^h)$ and $\Delta H_m^h$ values as described above. These values at different [P]/[N] ratios are shown in Table 1. The intermediate phase of the transition shown in Figure 1B were analyzed for $\Delta G_0(T)$, the Gibbs free energy change associated with the melting of the triplex, $h_1^\text{s}$, using the following equation,

$$\Delta G_0(T) = -RT \ln \left[ \frac{C_o(y(T)-y_a(T))^2}{(y_a(T)-y(T))(y(T)-y(T))} \right]$$  \hspace{1cm} (3)$$

where $C_o$ is the total strand concentration, $y(T)$ is the observed optical property for the intermediate phase at T K, $y_a$ is the optical property of the single strand when it is dissociated from the triplex (i.e., in the post transition region of the melting of the triplex) and shows a small temperature-dependence, $y_i$ is the optical property of the triplex (i.e., the complex of hairpin in the native state and the single strand existing in the posttransition region of the first phase. $y_i$ also shows a small temperature-dependence. Stability curve ($\Delta G_0(T)$ versus T) were drawn at each [P]/[N] ratio as described above and analyzed for $T_m^d$ (midpoint of melting of triplex) and $\Delta H_m^d$ (enthalpy change at $T_m^d$). Values of $T_m^d$ and $\Delta H_m^d$ at different [P]/[N] ratios are given in Table 1.

**UV mixing curve determines the stoichiometry of interaction between RY12 and RY28**

If a single strand of RY12 combines with the hairpin RY28 to form a triplex structure, the stoichiometry of this interaction can be determined by UV mixing curve as shown in Figure 2. First order regression of the data points obtained in Job titration shows a clear inflection at 0.5 mole fraction in terms of strand concentration suggesting that complex formation occurs between one RY28 hairpin molecule and a single strand of RY12 in presence of REWER forming a three stranded structure.

**Gel retardation assay demonstrates the formation of triplex**

The autoradiogram of the gel-electrophoresis of the free RY12, RY28 and their 1:1 mixture in the presence and absence of the peptide REWER, are shown in Figure 3. Either of the oligomer was radiolabeled and all samples underwent same treatment...
before electrophoresis. A retarded band, indicative of the triplex is seen in the mixture of RY28 and RY12 only in the presence of the peptide. The same band is seen irrespective of whether RY12 or RY28 is labeled.

Fine mapping of triplex by chemical probing

KMnO₄ modifies exposed thymines at the C5–C6 double bond in single stranded or unwound region of nucleic acids [22], although purine modification has also been reported [23]. Results of KMnO₄ modification on RY28 hairpin are shown in Figure 4A (lane 1). Surprisingly all the thymines of RY28 are sensitive to KMnO₄ with T4, T6, T12, T14, and T15 being hypersensitive. The pattern is asymmetric and unusual and shows that the thymines of the 5'9-half are hypersensitive while those of 3'9-half are not. This pattern remains unchanged even after addition of RY12. Figure 4A, lane 2 shows the results of KMnO₄ modification on a 1:1 mixture of labeled RY28 and unlabeled RY12. The pattern for RY28 alone is drastically altered when REWER is present in the annealed sample; the 3'9-half of the hairpin becomes hypersensitive while 5'9-half is protected from KMnO₄ modification when the peptide is present in the sample. (Figure 4A, compare lane 3 with lanes 1 and 2). In the presence of REWER, 1:1 mixture of RY28 and RY12 gave a footprint of triplex (lane 4). Thymines are now resistant to KMnO₄ in the triplex, and protection is more pronounced at T26, T22, T20, T12 and T10. The overall reactivity pattern is consistent with A₅N₆-A-T and T₅N₆-A-T base triplets). Even thymines in the loop (T12 and T15) show little sensitivity. As in hairpin T4 is reactive in the triplex also possibly due to fraying ends.

DEPC is used as a DNA structural probe for detection of bases at B-Z junction [24] and at cruciform loop [25]. DEPC carboxylates purines at N7 [24]. Adenines in WC duplex show protection from DEPC modification when N7 is engaged in Hoogsteen pairing in normal antiparallel triplex [25].

Sample preparation

Table 1. Thermodynamic parameters associated with the melting transitions of the duplex, hairpin and triplex¹².

| Compound              | Tₘd  | ΔHₘd  | Tₘh  | ΔHₘh  | Tₘt  | ΔHₘt  |
|-----------------------|------|-------|------|-------|------|-------|
| RY12 (-Pep)³         | 35.0±0.1 | 45.0±2.0 |     |       | 68.0±0.2 | 75.0±5.0 |
| RY28 (-Pep)          |      |       |      |       |      |       |
| RY12+RY28 (-Pep)     | 35.1±0.1 | 45.0±3.0 |     |       | 68.8±0.1 | 80.3±2.0 |
| [P]/[N] = 15         | 35.3±0.2 | 45.6±1.0 | 54.5±0.1 | 142.0±9.0 | 69.5±0.1 | 96.8±2.0 |
| [P]/[N] = 30         | 35.7±0.1 | 52.5±2.0 | 53.7±0.1 | 148.0±8.0 | 69.9±0.1 | 98.0±3.0 |
| [P]/[N] = 45         | 32.1±0.1 | 43.7±1.0 | 52.5±0.1 | 98.9±3.0 | 65.5±0.1 | 89.0±1.0 |
| [P]/[N] = 60         | 29.9±0.4 | 42.0±1.0 | 50.3±0.2 | 102.0±4.0 | 65.1±0.1 | 83.0±1.0 |

¹Superscripts d, h and t represent duplex, hairpin and triplex, respectively.
²° gives the mean error of the triplicate measurements. The maximum standard errors of the least-squares analysis of the stability curve are ±0.5 °C and ±5 kcal mol⁻¹ for Tₘ and ΔHₘ, respectively.
³(-Pep) measured in the absence of the pentapeptide.

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Figure 2. UV mixing curves show the stoichiometry of interaction between RY28 and RY12 in the presence of REWER. Both the oligonucleotide solutions contained REWER at [P]/[N] ratio of 30 prior to mixing. The abscissa is in mole fraction of RY28. There is a clear inflection point at 0.5 mole fraction indicating that single strand of RY12 interact with RY28 hairpin.
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Figure 3. EMSA shows the formation of triplex between RY12 and RY28. Triple helix formation between dodecamer RY12 and RY28 hairpin in the presence of the peptide REWER detected on 15% native polyacrylamide gel electrophoresis. First four lanes show the results of experiments using radiolabeled RY12, last four lanes used radiolabeled RY28. * represents 5’ radiolabeled oligonucleotide.
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Peptide-Mediated Triplex Structure

Discussion

A prerequisite to an \textit{in vivo} parallel triple helix between a single strand of DNA and the homologous duplex DNA is the localized unwinding of DNA double helix. Previous studies have used a pentapeptide (KGWGK) where tryptophan intercalates between bases and induces unwinding of DNA [26]. REWER with a centrally placed tryptophan is suggested to do the same first by unfolding the DNA and then enabling the invasion of third strand resulting in the formation of triple helix.

The first indication of triplex formation comes from the analysis of UV thermal transition of 1:1 mixture of RY12 and RY28, in presence and absence of the pentapeptide REWER. The free oligomer RY12 in 100 mM NaCl exists as a duplex which melts cooperatively with $T_m$ and $\Delta H_m$ values of $40^\circ\mathrm{C}$ and $66 \text{ kcal mol}^{-1}$, respectively [26]. Under the present solution conditions the $T_m$ and $\Delta H_m$ for RY12 melts are $35\pm0.1^\circ\mathrm{C}$ and $45\pm2 \text{ kcal mol}^{-1}$, respectively. These lower values are possibly due to different buffer conditions used in the present study. The RY28 hairpin denaturation is also a cooperative process (Figure 1A). It is seen in Figure 1A that relative absorbance ($A_{rel}$) increases linearly up to $55^\circ\mathrm{C}$ followed by a sigmoidal transition and then a linear increase above $78^\circ\mathrm{C}$. We assume that the linear portions represent the pre-and posttransition region of a two-state, helix-coil transition. Analysis of the transition curve gave $T_m$ and $\Delta H_m$ values of $68\pm0.2^\circ\mathrm{C}$ and $75\pm5 \text{ kcal mol}^{-1}$, respectively, for the RY28 hairpin. On heating 1:1 mixture of RY28 and RY12, we observed a biphasic transition, where each curve represents the melting of individual oligonucleotide. Analysis of the biphasic curve gave values of $T_m$ and $\Delta H_m$ that are close to those of RY12 duplex and RY28 hairpin individually (Table 1), suggesting that RY12 and RY28 do not interact under these conditions. These results are in agreement with those reported earlier [13]. Upon addition of the pentapeptide REWER to the 1:1 mixture of RY12 and RY28, however, a triphasic transition was observed (Figure 1B). This additional phase transition appears in an intermediate temperature range, when the RY12 duplex is completely melted but the RY28 is intact. This raises the possibility of a single strand of RY12 interacting with the hairpin molecule to form a three-stranded structure with $T_m = 54.5^\circ\mathrm{C}$, which is incidentally higher than the $T_m = 35^\circ\mathrm{C}$ of normal antiparallel triplex formed between RY28 and YR12 [13]. Thermodynamic analysis of triphasic curve in the presence of different peptide concentrations gave values of $T_m$ and $\Delta H_m$ and these are included in Table 1. $T_m$ and $\Delta H_m$ values associated with denaturation of the duplex and the hairpin increased slightly after addition of the peptide up to a $[P]/[N]$ ratio of 30. A further addition of the peptide causes significant decrease in $T_m$ and $\Delta H_m$ (Table 1). One possible explanation is that the addition of a positively charged peptide at low concentration initially neutralizes phosphate charges and stabilizes the duplex structure. The stabilizing effect is offset at higher concentration of REWER, when the peptide binds in the major groove of the duplex or hairpin stem and destabilizes both the RY12 and RY28 by intercalating tryptophan side chain [26]. The thermodynamic characteristics of the purported triplex are given in Table 1. It is interesting to note that before destabilizing conditions are reached (i.e., when triplex has not fully formed) $\Delta H_m^a$ is the sum of $\Delta H_m^d$ and $\Delta H_m^h$. However, the values of $\Delta H_m^a$ at the two highest peptide concentrations, when peptide binding and intercalation is complete, are less than the sum of $\Delta H_m$ values of RY12 and RY28 suggesting an enthalpy driven complex formation between the two components.

![Figure 4. KMnO4 and DEPC sequencing results show protection of DNA bases due to triplex formation in the presence of REWER. Prior to electrophoresis all chemical modifications were done at 4°C. A) KMnO4 panel shows the reactivity of the free RY28 and RY28+RY12 (1:1) mixture with radiolabeled RY28 in the presence and absence of the pentapeptide, REWER. [P]/[N] ratio in all experiments was 3.8. Lane 1, RY28 alone; lane 2, RY28+R12 mixture; Lane 3, RY28+REWER; lane 4, RY28+RY12+REWER. Bands corresponding to thymines and adenines are indicated. B) DEPC reactivity of the triplex. Lane 1 represents probe only (without treatment) and was used as control for both KMnO4 and DEPC experiments. Lane 2, RY28; lane 3, mixture of radiolabeled RY12+RY28; lane 4, RY28+RY12+REWER. The bands corresponding to adenines and guanines are marked. *.represents 5' radiolabeled oligonucleotide. doi:10.1371/journal.pone.0065010.g004](image-url)
The stoichiometry of the interacting complex was determined by UV mixing curve in the presence of REWER. The inflection point at 0.5 mole fraction establishes the 1:1 stoichiometry of RY12 and RY28 oligomers in the complex indicating a three-stranded structure (Figure 2). Detection of the triplex on the native PAGE under standard conditions proved difficult due to small size and low overall stability of the complex. In order to encourage the triplex formation, samples were run in cold room with high concentration of the pentapeptide in the polyacrylamide gel itself. Under these conditions, 1:1 mixture of RY12 and RY28 showed a shifted band that may corresponds to triplex (Figure 3). The same shifted band is observed irrespective of whether RY28 or RY12 is radiolabeled, showing that the retarded band is not due to any structural artifact of either oligomer. The RY28 alone is not shifted even in the presence of the peptide.

The structure was finally probed with KMnO4 and DEPC modifications followed by sequencing. Ideally, chemical modification is performed on an isolated triplex, which in our study proved difficult due to the small size and low stability of the complex. We therefore carried out modification reaction on the annealed samples in the presence of the peptide. We are aware that during annealing various species are under equilibrium, and presence of such alternative structures will interfere with the band pattern of the RY28 hairpin molecule in isolation or in the triplex. We ensured that the association reaction between RY12 and RY28 was complete (as indicated by our thermal denaturation data) before precipitation and chemical treatments. It is reasonable to expect that under optimal experimental conditions the intense bands would be from modification of the major species of our interest and we may be able to detect a footprint for the triplex. Figure 4A shows the results of KMnO4 modification of several samples that were subjected to same treatment throughout. Unlike in EMSA we observed a complete protection of chemical groups in case of footprinting experiment suggesting the formation of stable triplex in presence of REWER. This discrepancy may be due to the fact that EMSA is a non-equilibrium method, whereas in footprint assay components of triplex are in binding equilibrium when treated with KMnO4 and DEPC [27]. Figure 4A, lane 1 shows result for RY28 hairpin alone. Due to their positions in loop of the RY28 hairpin T14, T15 and possibly T12 present at loop closure are expected to be modified by KMnO4. We observed that these bands are intense. Additionally prominent bands due to T4, T6 and T10 were also observed. The later three thymines on the 5'-arm of the hairpin are hypersensitive but the corresponding thymines on the 3'-arm are not. This asymmetric pattern can be explained if there are kinks at the junctions of purine and pyrimidine stretches exposing the thymines on one arm and burying the corresponding thymines on the opposite arm making them less sensitive [13]. Almost identical band pattern resulted upon KMnO4 modification of a 1:1 mixture of RY28 and RY12 (Figure 4A, lane 2) suggesting that they do not interact. KMnO4 sensitivity pattern of the hairpin molecule changes drastically in the presence of the pentapeptide, REWER (lane 3). The thymines on the 3'-arm of the hairpin are now hypersensitive while those on 5'-arm are not. We are unable to explain this observation except that it may be due to the binding of REWER to the hairpin. Following an earlier study with another peptide (KGWKG), REWER is expected to bind to RY12 duplex in the major groove resulting in the unwinding of the double helix by intercalating tryptophan [26]. When the triplex forms, i.e., with a 1:1 mixture of RY28 and RY12 in the presence of REWER, thymines on both the arms of RY28 becomes insensitive to KMnO4 and a footprint develops. Figure 4A (lane 4) shows that only T15 in the loop, T6 and T4 are sensitive to KMnO4.

Figure 5. Schematic representation of the proposed triplex structure. A) Base sequence of RY12 dodecamer. B) RY28 hairpin. C & D) Schematics of base pairing in the triplex formed between RY12 and RY28. It show the representation diagram of the two putative triplex formations via alternate strand recognition (−) represent Watson-Crick hydrogen bonds. The third strand is shown in lowercase. Open and solid arrows depict the polarity of third strand and hairpin duplex, respectively. Numbers 1–4 represent base triads in triplex structure. A parallel triplex is formed when three purines of the third strand form hydrogen bonds with underlying purine of the hairpin (PuPyPu) it then changes strand and binds to the other strand of the hairpin (PyPuPy) in an antiparallel orientation. Please note that unlike more commonly found orientation in PuPyPu triplex where chemically homologous strands show antiparallel polarity our model suggests homologous strands in parallel orientation. E) A three dimensional rendition for a triplex of type 3

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Consistent with the results of KMnO₄ modification, DEPC modification occurred on adenines and guanines of the RY28 hairpin (Figure 4B, lane 2). This pattern is not changed when DEPC modification is done on an annealed 1:1 mixture of RU12 and RY28 (Figure 4B, lane 3). It should be noted that the adenines, incidentally, are at the junctions of the stretches of purines and pyrimidines. DEPC is known to carboxylate bases at the B-Z junction [24], and at cruciform loops [25]. In the triplex, however, all the adenines are insensitive to DEPC and a triplex footprint is observed (Figure 4B, lane 4). This result although similar to that of KMnO₄ reaction, seemed surprising since the N7 of purines in a parallel triplex are expected to be free and sensitive to DEPC. In a previous study authors showed that with depurinated joint molecules from RecA recombination, the N7 of adenines and guanines on the anticomplementary strand were sensitive to DEPC and those on other strands remained insensitive [28]. In our proposed triplex structure, because of the palindromic nature of the sequence there is no specific anticomplementary strand and the observed bands due to A7, A9 and A17 are possibly due to contamination of minor species of malformed triplex as judged by the intensity of these bands when compared to the other lanes. It should be noted, however, that Chiu et al., used the depurinated molecule [29], which was possibly a collapsed triplex, whereas in the present study triplex is bound to the peptide, which may have blocked the N7 in the major groove. Thus the results of chemical probing indicate that the bases in our triplex are inaccessible to KMnO₄ or DEPC, which is consistent with the triplex structure proposed here. These results prompted us to conclude that RY28 hairpin joins with a single strand of homologous RU12 to form triplex helix. Figure 5C & D shows the schematic diagram of the two possible putative triplex formations via alternate strand recognition. In this complex, triplex involving PuPyPu type (Figure 5C&D triads 1&3) both purine strands have the same polarity and are parallel whereas triplex involving PyPuPy type have third strand in antiparallel orientation (Figure 5C&D triads 2&4). A shifted antiparallel triplex with a three base shift can also be conceived, but thermal stability data and chemical probing results as discussed below do not support such a structure.

The present investigation on possibility of triplex formation between RY28 hairpin and a single strand of RU12 in the presence of REWER, was prompted by an unusual observation of a triphasic melting curve of RU12 and RY28 in presence of REWER, which was not observed in the presence of KGWGGK [20]. We believe this difference in the behavior of the two peptides is due to their sequence difference as well as the differences in the extent of unwinding they produce by tryptophan intercalation. Amino acids side chain intercalation is an important parameter for homologous pairing. However, if the pentapeptide binds in the major or minor groove of DNA, invasion by the third strand to form the triplex must occur in the other unoccupied groove. Both the grooves thus being blocked, the bases of RY28 hairpin would be protected from KMnO₄ or DEPC modification resulting in footprint of the triplex as observed here. Present study do not conclusively prove the major or minor groove occupation of the third strand, it is interesting to note that our proposed triplex structure with tryptophan intercalation every three bps [26] is strikingly similar to the theoretically proposed model of minor groove associated triplex [29]. Although peptide is essential for triplex structure we do not know the exact mechanism of REWER-induced triple-helix formation. We do not know if the binding of REWER and oligonucleotides is of covalent or noncovalent type. Structural studies on our well-defined system by NMR or X-ray diffraction only can resolve this issue. Our system serves not only as a model for studying ligand-mediated triplex formation but also has a potential to be used as tool in molecular biology and biochemistry [5,30,31].

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Author Contributions
Conceived and designed the experiments: TS FA KBVR. Performed the experiments: TS. Analyzed the data: AD TS FA VR. Contributed reagents/materials/analysis tools: KBVR. Wrote the paper: TS FA VR.

References
1. Gacy AM, Gottfluer GM, Spira C, Chen X, Gupta G, et al. (1998) GAA instability in Friedreich's Ataxia shares a common, DNA-directed and intrastrand mechanism with other trinucleotide diseases. Mol Cell 1: 593–599.
2. Bisler JJ (2007) Triplex DNA and human disease. Front Biosci 12: 4536–4546.
3. Casey BP, Glazer PM (2001) Gene targeting via triple-helix formation. Prog. Nucleic Acid Res Mol Biol 67: 163–192.
4. Panyunin IG, Neumann RD (2005) The potential for gene-targeted radiation therapy of cancers. Trends Biotechnol 23: 492–496.
5. van der Oost J (2013) Molecular biology. New tool for genome surgery. Science 339: 765–770.
6. Frank-Kamenetskii MD, Mirkin SM (1995) Triplex DNA structures. Annu Rev Biochem 64: 65–95.
7. Chen Z, Yang H, Pavletic N (2008) Mechanism of homologous recombination from the RecA-dependent DNA structures. Nature 453: 489–491.
8. Conway AB, Lynch TD, Zhang Y, Fortin GS, Fang CW, et al. (2004) Crystal structure of a Rad51 filament. Nat Structure Mol Biol 11: 791–796.
9. Laursen SL, Cox MM (2002) The bacterial RecA protein and the recombinational DNA repair of stalled replication forks. Annu Rev Biochem 71: 71–100.
10. Prevost C, Takahashi M (2003) Geometry of the DNA strands within the RecA-nucleoflament: role in homologous recombination. Q Rev Biophys 36: 429–453.
11. Zhurkin VB, Raghunathan G, Ulyanov NB, Camerini-Otero RD, Jernigan RL (1994) A parallel DNA triplex as a model for the intermediate in homologous recombination. J Mol Biol 239: 181–200.
12. van der Heijden T, Modesti M, Hage S, Kanaar R, Wyman C, et al. (2008) Homologous recombination in real time: DNA strand exchange by RecA. Mol Biol Cell 30: 530–538.
13. Sahi T, Roy KB. (2000) Doulecamer d-AGATC TAGATC And A Homologous Hairpin Oligomer Form a Novel Intermolecular Triplex. J of Biochem Mol Biol 339: 768–770.
14. Rubin CM, Schmid CW (1980) Pyrimidine-specific chemical reactions useful for DNA sequencing. Nucleic Acids Res 8: 4613–4619.
15. Rishi V, Bhattacharya P, Chatterjee R, Rosenberg J, Zhao J, et al. (2010) CpG methylation of half-CRE sequences creates C/EBPalpha binding sites that activate some tissue-specific genes. Proc Natl Acad Sci U S A 107: 20311–20316.
16. Banerjee A, Bose HS, Roy KB (1991) Fast and simple anion-exchange chromatography for large-scale purification of self-complementary oligonucleotides. Biotechniques 11: 650–656.
17. Plum GE, Park YW, Singleton SF, Dervan PB, Breslauer KJ (1990) Thermodynamic characterization of the stability and the melting behavior of a DNA triplex: a spectroscopic and calorimetric study. Proc Natl Acad Sci U S A 87: 9436–9440.
18. Rishi V, Bhattacharya P, Chatterjee R, Rosenberg J, Zhao J, et al. (2010) CpG methylation of half-CRE sequences creates C/EBPalpha binding sites that activate some tissue-specific genes. Proc Natl Acad Sci U S A 107: 20311–20316.
19. Rubin CM, Schmid CW (1980) Pyrimidine-specific chemical reactions useful for DNA sequencing. Nucleic Acids Res 8: 4613–4619.
20. Williamson JR, Celander DW (1990) Rapid procedure for chemical sequencing of small oligonucleotides without ethanol precipitation. Nucleic Acids Res 18: 379.
21. Taneja S, Ahmad F (1994) Increased thermal stability of proteins in the presence of amino acids. Biochimica et Biophys Acta 1170: 147–153.
22. Glover JN, Farah CS, Pulleyblank DE (1990) Structural characterization of separated H DNA conformers. Biochemistry 29: 11110–11115.
23. Rao BJ, Radding CM (1993) Homologous recognition promoted by RecA protein via non-Watson-Crick bonds between identical DNA strands. Proc Natl Acad Sci U S A 90: 6646–6650.
24. Johnston BH, Rich A (1985) Chemical probes of DNA conformation: detection of Z-DNA at nucleotide resolution. Cell 42: 713–724.
25. Furlong JC, Lilley DM (1986) Highly selective chemical modification of cruciform loops by diethyl pyrocarbonate. Nucleic Acids Res 14: 3995–4007.
26. Roy KB, Kaikerti S, Bose HS, Chauhan VS, Rajeswari MR (1992) Hairpin and duplex forms of a self-complementary dodecamer, d-AGATCTAGATCT, and interaction of the duplex form with the peptide KGWGK: can a pentapeptide destabilize DNA? Biochemistry 31: 6241–6245.
27. Hellman LM, Fried MG (2007) Electrophoretic mobility shift assay (EMSA) for detecting protein-nucleic acid interactions. Nat Protoc 2: 1849–1861.
28. Chiu SK, Rao BJ, Story RM, Radding CM (1993) Interactions of three strands in joints made by RecA protein. Biochemistry 32: 13146–13153.
29. Bertucat G, Lavery R, Prevost C (1998) A model for parallel triple helix formation by RecA: single-single association with a homologous duplex via the minor groove. J Biomol Struct Dyn 16: 535–546.
30. Washbrook E, Fox KR (1994) Alternate-strand DNA triple-helix formation using short acridine-linked oligonucleotides. Biochem J 301 (Pt 2): 569–575.
31. Duca M, Vekhoff P, Oussedik K, Halby L, Arimondo PB (2008) The triple helix: 50 years later, the outcome. Nucleic Acids Res 36: 5123–5138.