Plasmonic photoluminescence for recovering native chemical information from surface-enhanced Raman scattering

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Surface-enhanced Raman scattering (SERS) spectroscopy has attracted tremendous interests as a highly sensitive label-free tool. The local field produced by the excitation of localized surface plasmon resonances (LSPRs) dominates the overall enhancement of SERS. Such an electromagnetic enhancement is unfortunately accompanied by a strong modification in the relative intensity of the original Raman spectra, which highly distorts spectral features providing chemical information. Here we propose a robust method to retrieve the fingerprint of intrinsic chemical information from the SERS spectra. The method is established based on the finding that the SERS background originates from the LSPR-modulated photoluminescence, which contains the local field information shared also by SERS. We validate this concept of retrieval of intrinsic fingerprint information in well controlled single metallic nanoantennas of varying aspect ratios. We further demonstrate its unambiguity and generality in more complicated systems of tip-enhanced Raman spectroscopy (TERS) and SERS of silver nanoaggregates.
Localized surface plasmon resonances (LSPRs) in metallic nanostructures can confine incident light into a nanoscale volume characterized by strongly enhanced optical fields. These hot-spots can dramatically enhance the Raman scattering, fluorescence, infrared absorption, or nonlinear optical signal, leading to a family of plasmon-enhanced spectroscopies. These techniques have become powerful tools for identification and characterization of molecular species related to chemistry, energy, material sciences, and physics and for analysis of bio-related systems, owing to their extraordinary sensitivity reaching single molecule level.

Notably, the relative intensity of Raman peaks in both surface-enhanced Raman scattering (SERS) and tip-enhanced Raman scattering (TERS) has been widely used to investigate the surface configuration of adsorbed molecules, charge transfer mechanism, local temperature, vibrational pumping, plasmon-mediated photocatalysis, and local chemical properties. However, it was frequently observed that the relative intensity of different Raman peaks may vary from spot to spot even on the same surface, which is unexpected from the chemical interaction point of view because the chemical properties should remain similar. Importantly, LSRRs do not only provide a large electromagnetic enhancement but also induce strong modifications of the relative strength of the peaks in the SERS spectra. The latter effect can be seen as a plasmonic spectral shaping effect (PSSE). The distorted spectral features as a result of the PSSE can severely undermine the interpretation of the experimental data when extracting chemical information like molecular orientation, the charge transfer mechanism in plasmon-mediated photocatalysis, local chemical properties, and so forth. It is thus desirable to reach a deeper understanding of the PSSE and to develop a method to recover the native chemical information.

The plasmon-induced PSSE was studied by Itoh et al., who successfully demonstrated a correlation between the relative strength of the SERS peaks of the molecules and the scattering of the SERS substrate. The first reported method to directly correct the PSSE was successfully demonstrated by Buchanan et al. using what they identified as the incompletely quenched surface-enhanced fluorescence (SEF) signal of dye molecules on SERS substrates. An advantage of this method is that the SEF is directly affected by the near field response, as in the case for the PSSE. However, this method is not applicable to more general systems in the absence of dye molecules or with strongly quenched fluorescence. Due to these limitations, no robust and systematic procedure has been broadly adopted by the SERS community to treat the PSSE with generality. As a result, information from ‘untreated’ Raman spectra is still constantly used as the basis to reach incorrect conclusions regarding selection rules in vibrational spectroscopy, a situation detrimental for further developments of SERS, and even of other fields related to plasmon-enhanced chemical reactivity or two-dimensional materials spectroscopy. Thus, it is important to develop a general method to correct for the PSSE that is robust and universally adoptable.

The photoluminescence (PL) of metal nanostructures has been connected to the behavior of the near fields inside the nanoparticle and has been considered to contribute to the broad ‘continuum’ background in typical SERS spectra. This background is commonly neglected or just directly subtracted to obtain a clean SERS spectrum, but its origin as PL means that it should contain similar near field information as the PSSE. Therefore, the SERS background may potentially be used to correct the PSSE and to recover the intrinsic chemical information contained in the relative strength of the Raman peaks.

Here we use single-gold nanorods as nanoantennas, which are resonant at different wavelengths but show the same chemical properties, to demonstrate such concept of correction. We carefully correlate their morphology, LSPR scattering, PL, SERS background and relative intensity of the SERS peaks (Fig. 1). We identify the contribution from the material response and the plasmonic enhancement to the PL through measuring the scattering and the PL response of the individual nanorods. We further quantitatively connect the PL and SERS background by measuring the PL and SERS of single nanorods before and after the adsorption of molecules, respectively. Based on these results, we propose a robust method to correct the PSSE that consists in normalizing the SERS by the SERS background. We successfully apply this approach to resolve and understand the PSSE signal in nanorods system, and corroborate the results by analytical models and numerical simulations. The generality of the proposed method is further supported by the results from more complex plasmonic systems including gap-mode TERS configurations and silver nanoaggregates.

**Results**

**Correlating the PL with the LSPR response.** We first connect the morphology of the individual nanorods and their single-nanoparticle optical response (scattering and PL) using a quartz substrate with microfabricated markers (Supplementary Fig. 1) for co-localization between the dark-field microscope and the scanning electron microscope (SEM). We choose a first set of five gold nanorods with different aspect ratios for the following experiments, which are marked in the dark-field image (Fig. 2a) and characterized by SEM (Fig. 2b). The PL (solid lines in Fig. 2c) and scattering (dotted lines in Fig. 2c) spectra of the five nanorods were excited with a 633 nm continuous wave (CW) laser and a white source, respectively, and were collected through the same objective using a home-built combined dark-field microscope and Raman microscope system (See schematic diagram of the system in Supplementary Fig. 2). To increase the signal-to-noise ratio, the polarization of both the laser and white light were kept parallel to the longitudinal axis of the nanorods in all measurements. We observe from Fig. 2c that, similarly to what has been reported in the literature, the PL spectra of the five gold nanorods show a prominent peak (in the 633 nm–720 nm range) that roughly follows the LSPR scattering spectra. However, the PL peaks are blue-shifted with respect to the scattering peaks and the shifts

![Figure 1](Image) **Figure 1 | Schematic diagram for the quantitative experimental study.** To correlate the LSPR, PL and SERS of single nanorods, the dark-field scattering and the PL signal are first measured in the absence of molecules, and SERS spectra were then measured after molecules adsorption.
depend on the resonant energy, becoming larger for the nanorods of larger aspect ratio that resonate at longer wavelengths.

The physical origin of PL of gold nanostructures is still under debate\textsuperscript{[30,31,43,44,46]}, but we consider a simple model that explains our PL measurements without invoking the detailed mechanism by which the energy is dissipated. Thus, by PL we mean here the emission at a smaller energy of the incident laser. We treat the PL as an incoherent process (phases are ignored) initiated by the incoming light of amplitude $E_0$ and angular frequency $\omega_{\text{exc}}$ polarized along the nanorod axis. The excitation efficiency is proportional to the local field intensity $|E_{\text{ins}}(r,\omega_{\text{exc}})|^2$ at each position $r$, where $E_{\text{ins}}(r,\omega_{\text{exc}})$ is the field induced inside the nanostuctures. During the PL processes, optical dipoles (which can also be seen as currents oscillating at optical frequencies) are excited at the emission angular frequency $\omega_{\text{em}}$ (energy loss $\hbar(\omega_{\text{exc}} - \omega_{\text{em}})$) with a probability $\eta(\omega_{\text{exc}},\omega_{\text{em}})$, and emit at a rate $I_{\text{em}}(r,\omega_{\text{em}})$. We thus obtain a local PL signal proportional to $|E_{\text{ins}}(r,\omega_{\text{exc}})|^2I_{\text{em}}(r,\omega_{\text{em}})$.

Using reciprocity\textsuperscript{[31,42]} to relate the emission rate from a dipole to the local field enhancement at the same angular frequency (see Supplementary Notes 1 and 2, and Supplementary Figs 4 and 5), we write the emission rate as $I_{\text{em}}(r,\omega_{\text{em}}) \propto |E_{\text{ins}}(r,\omega_{\text{em}})/E_0|^2$, where $E_{\text{ins}}(r,\omega_{\text{em}})$ is the local field induced by a plane wave of angular frequency $\omega_{\text{em}}$ polarized along the axis of the nanorod (equivalent to $E_{\text{ins}}(r,\omega_{\text{exc}})$ but for the emission frequency). The $\omega_{\text{em}}^4$ factor describes the frequency-dependence of the emission from a dipole of a fixed strength. After integrating over the volume $V$ of the nanoparticle, we obtain an expression, consistent with previous work\textsuperscript{[31,33]},

$$\text{PL}(\omega_{\text{exc}},\omega_{\text{em}}) \propto \int \omega_{\text{em}}^4 \frac{|E_{\text{ins}}(r,\omega_{\text{exc}})|^2}{E_0} \frac{|E_{\text{ins}}(r,\omega_{\text{em}})|^2}{E_0} I_{\text{em}}(\omega_{\text{exc}},\omega_{\text{em}}) dV,$$  

where PL($\omega_{\text{exc}},\omega_{\text{em}}$) is the photoluminescence emission at $\omega_{\text{em}}$ for excitation at $\omega_{\text{exc}}$ and $I_{\text{em}} \propto E_0^2$ is the incident intensity.

The dependence of the PL on the local fields will be a key below when discussing the method to correct the PSSE on the SERS signal, but it makes the direct verification of equation (1) more difficult, because $E_{\text{ins}}(r,\omega_{\text{exc}})$ is not easily accessible in experiments. Fortunately, for simple systems where a single resonance dominates the response and the radiative damping is low\textsuperscript{[43]}, the scattering $S(\omega_{\text{em}})$ and the square of the local fields $|E_{\text{ins}}(r,\omega_{\text{em}})/E_0|^2$ exhibit a similar energy dependency. Under this assumption, the PL can be expected to depend on the emission frequency approximately as:

$$\text{PL}(\omega_{\text{em}}) \propto \omega_{\text{em}}^4 \eta(\omega_{\text{exc}},\omega_{\text{em}}) S(\omega_{\text{em}}) \propto \text{PL}_{\text{bulk}}(\omega_{\text{exc}},\omega_{\text{em}}) S(\omega_{\text{em}}),$$

where $\text{PL}_{\text{bulk}}(\omega_{\text{exc}},\omega_{\text{em}}) \propto \omega_{\text{em}}^4 \eta(\omega_{\text{exc}},\omega_{\text{em}})$ captures the bulk properties of the material. As a consequence, we should obtain the same $\omega_{\text{em}}$-dependence of the ratio between photoluminescence and scattering $R^\text{PL} = \text{PL}/S$ when comparing different nanostructures. This expectation is confirmed experimentally, as shown in the results at the bottom right of Fig. 2c, where we show an almost identical $R^\text{PL}$ for the five different nanorods under study. Furthermore, we also measured the PL of an atomically smooth gold(111) single-crystal surface (PL$_{\text{bulk}}$, blue line in bottom right of Fig. 2c), as essentially corresponding to the bulk gold PL response. Indeed, the spectral shape for this measurement is almost identical to the $R^\text{PL}$ ratio from the nanorods, in agreement with the description of the PL obtained in equation (2). Nonetheless, we emphasize that, in general, the PL should depend on the near field as given by equation (1), and that the use of $S(\omega_{\text{em}})$ in equation (2) and Fig. 2c would only be valid for particular cases. The derivation of equations (1) and (2), and the conditions where the different approximations are valid are discussed in more detail in Supplementary Notes 2 and 4, and Supplementary Figs 5 and 8 (particularly how some of the relatively small, higher order corrections partially cancel, giving an almost perfect agreement in the bottom right panel of Fig. 2c).

Quantitative relationship between PL and the SERS background.

Since PL is due to the intrinsic response of the nanorods, we expect it to appear as a background in the SERS spectra\textsuperscript{[31,34,35]}. To assess this connection quantitatively, we successively measure the PL (dashed lines in Fig. 3a–e) and SERS$^{\text{raw}}$ signal (solid lines in Fig. 3a–e) of another set of five nanorods of different aspect ratio (SEM images in the insets of Fig. 3) before and after molecule adsorption, respectively. We introduce
the superscript ‘raw’ to specify that we are discussing the Raman signal before subtracting the broad background. The Raman signal of non-resonant molecules adsorbed on a single Au nanorod is too weak to perform a reliable analysis. Therefore, we chose the functionalized malachite green isothiocyanate (MGITC) molecules (see Supplementary Fig. 3 for the molecular structure) as the target molecule, which has an absorption peak at about 630 nm, in resonance with the 633 nm excitation laser and thus ensuring a sufficiently high signal-to-noise ratio of single-nanoparticle SERS spectra. To avoid a strong fluorescence background from molecules adsorbed on the quartz substrate, the substrate was first modified by trichloro(1H, 1H, 2H, 2H-perfluorooctyl)silane (PFS) to impede the adsorption of the MGITC (see Methods for detailed experimental procedures). The use of the same molecule, nanorods synthesized in the same vessel and the same excitation laser strongly attenuate the possible contribution of different chemical effects to the SERS spectra.

Furthermore, the PL and SERS were measured on the same five nanorods, with the same optical set-up and under identical conditions (except for whether the molecules are present). The PL and SERS spectra in Fig. 3 can thus be directly and quantitatively compared, including the absolute strength. Crucially, we find that the SERS background, $B_g$, in presence of molecules closely resembles the PL spectrum before the molecules are deposited, in both spectral profile and strength. This similarity is observed for all the five nanorods with resonances at different frequencies, which clearly demonstrates that PL is the main contributor to the SERS background in the present system (this finding is further stressed by the more detailed analysis below); other possible contributions, such as the SEF of a residual amount of MGITC on the quartz substrate, are comparatively weak.

It can also be noted that there is a small red shift in the position of the SERS background with respect to that of the PL peak for all the five gold nanorods in Fig. 3a–e. We attribute the difference to the red shift of the LSPR induced by an increase of the refractive index outside the nanorod surface after adsorption of MGITC molecules. According to equation (2), the PL from simple systems is approximately proportional to the LSPR scattering. Therefore, to account for the effect of molecular adsorption on PL, we analyse the scattering spectra of the five nanorods before (dashed lines in Fig. 3f) and after (dotted lines in Fig. 3f) adsorption. After adsorption, the scattering spectra show significant red shift in wavelength and an increase in intensity, compared with that of bare gold nanorods, which agrees well with simulations of the optical response based on the finite-element method (Supplementary Figs 4 and 6). With the scattering spectra $S$ and the measured PL, PL(meas.) on hand, the corrected PL spectra, PL(corr.) can be written from equation (2) as

$$\frac{\text{PL(corr.)}}{S(\text{with molecules})} = \frac{\text{PL(meas.)}}{S(\text{without molecule})}. \quad (3)$$

![Image of the five nanorods in pseudo color, which show an increased aspect ratio from 100 nm to 100 nm.](Image)

![Table showing intensity (counts) vs. Raman shift (cm⁻¹) and wavelength (nm) for the five gold nanorods measured before and after adsorption of the MGITC molecule.](Table)

### Figure 3 | Quantitative correlation of the SERS background and PL for single gold nanorods.

(a–e) Measured PL signal, PL(meas.) (dashed lines), of five gold nanorods with different aspect ratio in the absence of molecules and the corresponding SERSraw spectra (which includes a spectrally broad background) after adsorption of the MGITC molecules (solid lines). All measurements were performed under 633 nm laser illumination. The dotted lines, PL(corr.), correspond to the PL after applying the correction described by equation (3), which accounts for the influence of the molecules on the plasmonic resonance. The insets are the SEM images of the five nanorods in pseudo color, which show an increased aspect ratio from 100 nm to 100 nm.

(f) Scattering spectra of the five gold nanorods measured before (dashed lines) and after (dotted lines) adsorption of the MGITC molecule.
Correcting the relative intensity of the SERS peaks using the PL. The understanding of the PL and its connection with the SERS background is used next to justify the main result of this paper, that is, how to suppress the PSSE from SERS measurements to retrieve intrinsic molecular fingerprint information. With this aim, we first plot in a single panel (Fig. 4a) all the SERS raw spectra shown separately in Fig. 3, corresponding to different aspect ratios of the nanorods and thus to different LSPR wavelengths. The relative strength of the peaks varies strongly from nanorod to nanorod. This point is further emphasized in Fig. 4b, where the background of the SERS spectra is first fitted and then subtracted (see Supplementary Note 5 for the detailed procedure) to obtain clean SERS spectra. In the following, we refer to the signal after background subtraction (SERS\textsubscript{sub} = SERS\textsuperscript{raw} - Bg), which allows to focus on the narrow Raman peaks. For Nanorod 1, the LSPR peak is at around 648.2 nm and the Raman peak corresponding to a 438.7 cm\textsuperscript{-1} Raman shift (emission at 651.1 nm) has the highest intensity, with the Raman peak for 1,614.5 cm\textsuperscript{-1} (705.1 nm) exhibiting a much lower intensity. In contrast, for Nanorod 5 supporting a LSPR at 708.9 nm, the 1,614.5 cm\textsuperscript{-1} (705.1 nm) Raman peak becomes the strongest and the Raman peaks for 205.2 cm\textsuperscript{-1} (641.3 nm) and 438.7 cm\textsuperscript{-1} (651.1 nm) are significantly weaker. As previously introduced, we discard changes on the chemical interaction between molecules and nanorods as an explanation of such a drastic change in the relative SERS intensity, because all the nanorods were synthesized in the same flask and thus chemical interactions should be the same.

In contrast, the PSSE provides a robust understanding of the observed changes because, for all the five nanorods, the Raman peaks are more clearly enhanced when their corresponding emission energy is close to the peak position of the LSPR spectra. It is apparent from Fig. 4a,b that the PSSE can be very significant, and it is crucial to consider it for a reliable interpretation of the measurements. For example, it has been pointed out by Itoh et al.\textsuperscript{21} that the temperature of a molecule could be erroneously estimated if the relative intensity of the anti-Stokes and Stokes lines are directly compared without considering the plasmonic response, up to the point of inferring a temperature that exceeds the decomposition temperature of the target molecule.

We argue next that the spectral behavior of the PL (and thus of the SERS background) and of the PSSE should resemble each other because both depend on the near fields in a similar manner, which facilitates disentangling the chemical information from the electromagnetic contribution to SERS. The electromagnetic Raman enhancement is proportional to both the local intensity enhancement at the illumination frequency (excitation) and the
radiation enhancement (emission)\textsuperscript{44}. As in the discussion of the PL, due to reciprocity, the latter is typically proportional to the local intensity enhancement at the emission frequency. The intensity of the SERS peaks from the different vibrational modes of the molecule can then be expressed as

$$\text{SERS}(\omega_{em}, \omega_{cm}) \propto \int E_0 \frac{E_{\text{out}}(r, \omega_{cm})}{E_0} \left| \frac{E_{\text{out}}(r, \omega_{cm})}{E_0} \right|^2 \sigma_{\text{Raman}} \mathrm{d}V,$$  

(4)

where $\sigma_{\text{Raman}}$ is the Raman cross section of the molecule containing the chemical information and $E_{\text{out}}(r, \omega_{cm})/E_0$ are the local field enhancement factor at the excitation and emission frequencies, respectively, evaluated at the position of the molecules, that is, outside the nanoparticle. The PSSE in our experiment would be a consequence of the dependence of the local near fields on $\omega_{cm}$ in equation (4).

The near fields are not easily accessible to experiments\textsuperscript{45,46}, which makes the direct evaluation of equation (4) to extract $\sigma_{\text{Raman}}$ a considerable challenge. A first alternative is to follow a similar approach as used in equation (2), and to consider the scattering $S(\omega_{cm})$ at the emission frequency to be roughly proportional to the square of corresponding near fields (see Supplementary Fig. 7b). We can simplify equation (4) to

$$\text{SERS}(\omega_{em}) \propto \sigma_{\text{Raman}} S(\omega_{em}).$$  

(5)

Under adequate conditions, this simple relationship can be sufficient to reveal the relative value of $\sigma_{\text{Raman}}$ for the different Raman peaks, as experimentally observed by Itoh \textit{et al.}\textsuperscript{21,27}. We experimentally corroborate that we can effectively suppress the PSSE for the nanorods by dividing the background-subtracted SERS\textsubscript{sub} spectra in Fig. 4b by the corresponding scattering spectra (dotted lines in Fig. 3f). The resulting SERS\textsubscript{sub} ($\omega_{em}$)/S($\omega_{em}$) spectra (Fig. 4c) is indeed very similar for all the five nanorods of different aspect ratio, in clear contrast with the large differences before the correction (Fig. 4b). Moreover, the five normalized spectra are almost identical to the Raman signal of the same molecules adsorbed on a gold(111) single-crystal surface (shown in the figure by the bottom blue line), further evidence that the normalized results indeed correspond to $\sigma_{\text{Raman}}$.

For well-defined dipolar modes as those presented here, the normalization procedure suggested by equation (5) appears as reliable and robust. However, we do not expect equation (5) to be valid with generality, because the connection between the near fields and the scattering can become very complex in a general nanoantenna configuration. These more challenging scenarios can include, for example, structures that support several plasmonic modes at the relevant frequencies, or systems whose far field is dominated by a non-resonant background unrelated to the modes dominating the near field. Furthermore, scattering spectra can be difficult to obtain in typical SERS and TERS experiments without integrated dark-field microscopy in the Raman system.

To circumvent these difficulties, it is convenient to compare the equations that show explicitly the dependence of the PL signal and the Raman enhancement on the near fields (equation (1) and equation (4)). Identifying the SERS background, $B_g$, as the PL (Fig. 3), using $\text{PL}_{\text{bulk}}(\omega_{em}, \omega_{cm}) \propto \omega_{em}^4 \eta(\omega_{em}, \omega_{cm})$, and assuming that the integrals over the local field intensity inside (equation (1)) or outside (equation (4)) the metal nanostructure exhibit a similar $\omega_{em}$-dependence (see Supplementary Notes 3,4 and Supplementary Fig. 9 for a more detailed discussion of the derivation and approximations involved), we obtain

$$\text{SERS}(\omega_{em}) \propto \frac{S(\omega_{em})}{\text{PL}(\omega_{em})},$$  

(6)

and the intrinsic Raman signature of the molecule can thus be written as

$$\sigma_{\text{Raman}}(\omega_{em}) \propto \frac{\text{SERS}(\omega_{em})\text{PL}_{\text{bulk}}}{B_g}. $$  

(7)

We evaluate equation (7) by dividing the background-subtracted SERS\textsubscript{sub} spectrum of each nanorod by the corresponding SERS background and then multiplying by the previously obtained PL\textsubscript{bulk} (PL measured for a gold(111) single crystal, Fig. 2c), SERS\textsubscript{sub} × PL\textsubscript{bulk}/B_g. The detailed data processing procedure, and the comparison between the fitted background and the PL of the same nanorods before molecule adsorption are provided in the Supplementary Note 5 and Supplementary Fig. 10. The obtained corrected Raman spectra are again almost identical for all the five nanorods (Fig. 4d), and very similar to the spectra measured for the molecules on a gold(111) surface (blue line). Thus, the correction scheme given by equation (7) behaves as intended for retrieving the intrinsic information from the molecule.

It is interesting to compare our approach with that reported by Buchanan \textit{et al.}\textsuperscript{28}, who directly normalize the SERS spectra with the SERS background. We emphasize that they identified their broad background in the SERS experiments not as PL from the metal but as SEF from the incompletely quenched dye molecules. The SEF would thus play a similar role in their work as the PL in our results. However, if we use the background to directly correct the spectra (Fig. 4e), that is, without the PL\textsubscript{bulk} factor, we can identify a noticeable difference in the relative intensity for the two peaks at 641.3 nm (Raman shift 205 cm\textsuperscript{-1}) and 705.1 nm (1,614 cm\textsuperscript{-1}; Fig. 4e), when comparing the results for the nanorods with those for the gold single-crystal surface. Thus, the correction of the PSSE to recover the intrinsic Raman spectra is more robust and systematic when we use the full equation (7), including the intrinsic PL of the gold.

Generalization to TERS and more complicated SERS systems. Equation (7) is obtained by relating two magnitudes, the PL and the SERS enhancement, both of which directly depend on the near fields. As a consequence, calculating SERS\textsubscript{sub} × PL\textsubscript{bulk}/B_g should also allow to recover the intrinsic chemical information for rather general systems. To evaluate the validity of our understanding and the generality of the correction method, we further applied it to tip-substrate (gap-mode TERS) and silver nanoaggregate systems.

As shown in Fig. 5a, two TERS spectra were obtained with two different gold tips on the same one-atom-thick palladium layer deposited on a gold(111) single-crystal surface adsorbed with 4-chlorophenyl isocyanide molecules. In this case, the illumination at 633 nm is off-resonant with the electronic transitions of the molecule. Both the background and the relative intensity of the main Raman peaks of the two TERS spectra are quite different. In the case of tip1, the intensity of the Raman peak at 1,168 cm\textsuperscript{-1} is slightly stronger than that at 1,586 cm\textsuperscript{-1} (after background subtraction, see inset). Inversely, for tip2 the intensity of the Raman peak at 1,168 cm\textsuperscript{-1} is significantly weaker than that at 1,586 cm\textsuperscript{-1}. In TERS, the tip is not in contact with the molecule, and the interaction of the molecule with the substrate should be the same for the two measurements, so that we safely attribute the changes to differences on the plasmonic response. Following the correction method introduced in Fig. 4d, by dividing the background-subtracted TERS spectra by their own fitted backgrounds (Fig. 5a, dashed lines) and multiplying by the bulk gold PL (Fig. 2c, blue line), that is, TERS\textsubscript{sub} × PL\textsubscript{bulk}/B_g, we obtain almost the same spectrum for the two tips (Fig. 5b).

Last, we consider a canonical SERS substrate such as that composed of silver nanoaggregates made of chemical synthesized
silver colloids. These complex plasmonic systems are widely used SERS substrates and allow to test the correction method for a different material showing complex couplings between the constituent nanoparticles. Three typical SERS\textsuperscript{raw} spectra of Rhodamine 6G molecules on the silver nanoaggregates excited at 633 nm (off-resonance) are shown in Fig. 5c, which again exhibit different relative intensity of the Raman peaks. By dividing the three background-subtracted SERS spectra by their own fitted background (Fig. 5c, dashed lines) and multiplying by the bulk silver PL, SERS\textsuperscript{sub} × PL\textsubscript{bulk}/Bg, we obtain three spectra (Fig. 5d) with remarkably similar relative intensity of the different Raman peaks, further validating the generality of the proposed method for retrieving intrinsic fingerprint information.

Discussion

In conclusion, we have proposed a general approach to correct the influence of the plasmon resonance dispersion on the relative intensity of the SERS peaks, known as the PSSE, to retrieve intrinsic information of the Raman signal of molecules. Our intrinsic fingerprint retrieval method exploits the PL-induced SERS background as an intrinsic spectral feature (also an internal standard) that contains the same near field information as the PSSE, and allows to recover molecular fingerprints that reflect the intrinsic chemical interaction of the molecule with the substrate. From a practical perspective, once the PL\textsubscript{bulk} is known, which should only depend on the material and the illumination wavelength, our method can be applied to standard SERS or TERS experiments, without requiring any additional measurements. In particular, as our procedure does not rely on the scattering signal, it can be applied to situations where the far field signal is difficult to measure or is uncorrelated to the near fields, that is, to the SERS enhancement. After correcting the raw Raman signal with the method proposed here, the corrected spectra may provide more reliable starting point to explore complex phenomena, related to the molecule orientation, to changes on the selection rules due to the interaction with the metallic particle, to vibrational pumping or other strong field effects, or to very large field gradients. This method can also serve as a powerful tool to clarify the origin of the photoluminescence, an important but yet controversial issue\textsuperscript{47}.

The intrinsic fingerprint retrieval procedure seems particularly necessary for systems that are expected to show a strong PSSE, such as SERS substrates with narrow and well-defined LSPR, and SERS of single or few molecules that only probes a certain hot spot with narrow LSPR mode at a time. In contrast, for many molecules over highly disordered SERS substrates, the PSSE could smear out and the correction may become less relevant. We have successfully applied the method to resonant and non-resonant Raman molecules, different plasmonic materials and three typical systems: single nanorods (as a well-defined model system), TERS and nanoaggregates (of direct relevance to practical application). A general protocol to correct the PSSE as presented here could be very useful in analytical sciences, and opens a new avenue towards unambiguous retrieval of native chemical information from the relatively complex and distorted information of Raman peaks in SERS and TERS spectra.

Methods

Sample preparation. We synthesized the gold nanorods following a reported binary surfactant method\textsuperscript{48}. The gold nanoseeds were synthesized first and a growth process was performed 30 min afterwards. The nanoseeds were synthesized
by injecting freshly prepared sodium borohydride (NaBH₄) (0.6 ml, 0.01 M) into the H₂AuCl₄·hexasolventtrimethylammonium bromide (CTAB) mixture (H₂AuCl₄ 5 ml, 0.01 M; CTAB: 5 ml, 0.2 M) under vigorous stirring in 30 °C water bath. The growth process can be divided into three steps. First, the growth solution was prepared by mixing the H₂AuCl₄ solution (25 ml, 1 mM), the silver nitrate (AgNO₃) solution (1.8 ml, 4 mM) with the CTAB–sodium oleate (NaOL) solution (25 ml Milli-Q water; CTAB: 0.90 g; NaOL: 0.15 g) in 50 ml cuvette. Second, the hydrochloric acid (HCl) (150 μl, 37 wt. % in water) was added to tune the pH. Finally, the ascorbic acid (125 μl, 64 mM) was injected into the solution to reduce the Au(III) (yellow) into Au(I) (colorless) and finally the nanoseeds solution (200 μl) was added into the solution for growth. The cuvette was kept in 30 °C water bath overnight after gently shaken. The average LSPR peak of nanorods can be tuned by adjusting the amount of CTAB, NaOL, AgNO₃ and HCl, accordingly. All the nanorods used in this study were from the same cuvette to make sure they were in exactly the same chemical environment. The innate size variability provides nanodisks with different aspect ratios. The nanoparticles were centrifuged and washed three times with Milli-Q water to remove the CTAB and NaOL. After being diluted for 100 times, the nanorods solution was dropped onto a quartz substrate with co-localization markers. To dry the sample, the substrate was put into a vacuum desiccator connected to a water circulation vacuum pump.

To correlate the SERS and PL spectrum of single nanorods with and without molecules, it is important to allow the molecule to adsorb only on the nanorod surface other than the quartz surface. Therefore, the surface of the quartz substrate was modified with special functional groups to prevent non-specific binding of the target fluorescence molecule. The modification was carried out by chemical vapor deposition of trichloro(1H,1H,2H-perfluorooctyl)silsesquioxane (PFS) from Aldrich. The quartz substrate was placed in a sealed vessel. A few drops of PFS was dispersed on the bottom of the vessel. The vessel was kept in an oven at 80 °C for 6 h to make sure the vapor of PFS fully reacted with the OH groups on the quartz surface. After the PL measurement, the quartz substrate was immersed in the solution containing the modified MGITC molecule for 1 h. Then, the sample was thoroughly rinsed by copious amount of Milli-Q water and immersed in ethanol for more than 1 h before further Raman measurements.

Silver nanoparticle synthesis was followed by the Lee–Meisel citrate reduction method. After twice centrifuging and washing, an aliquot of silver nanoparticle solution was incubated with 1 ml of 1 mM NaCl for 2 h. The diameter of the silver nanoparticles was about 15 nm. Silver nanostructures were demonstrated to show a nearly identical feature at different polarization angles. The scattering spectra were recorded with variable polarization angles relative to the long axis of the nanorod, the normalized SERS spectra show the slight polarization dependence due to the 3 nm Pt deposition, the total length as well as the diameter of the nanorod were slightly adjusted to match the measured dark-field scattering spectra. See Supplementary Note 1 for detailed simulation parameters.

**Data availability.** The data that support the findings of this study are available from the corresponding authors upon reasonable request.

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Acknowledgements

We acknowledge helpful discussions with Jeremy Baumberg. We acknowledge support from NSFC (21633065, 21621091 and J1310024), MOST (2013CB933703 and 2016YFA0200601), MOE (IRT13036), and MINECO (FIS2016-80174-P) and Basque Government grant of UPV/EHU Grupos Consolidados IT-756-13. R.E. also acknowledges fellow Gipuzkoas del Gipuzkoako Foru Aldundia through the European Union’s FEDER program ‘Una manera de hacer Europa’.

Author contributions

Experiments were designed by K.-Q.L., X.W. and B.R. The experiments and data processing were executed by K.-Q.L. FEM simulations were performed by J.Y. and the TERS experiment were conducted by J.-H.Z. Quartz substrate with markers was micro-fabricated by J.-Y.L. Modification of quartz substrates was helped by S.H. Instrument calibration was supported by B.-J.L. SEM characterization was helped by Z.-C.L. The data processing was supported by Z.-C.L. The theoretical analysis and data interpretation were performed by K.-Q.L., J.Y., X.W., B.R., R.E. and J.A. The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

Additional information

Supplementary Information accompanies this paper at http://www.nature.com/naturecommunications

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How to cite this article: Lin, K.-Q. et al. Plasmonic photoluminescence for recovering native chemical information from surface-enhanced Raman scattering. Nat. Commun. 8, 14891 doi: 10.1038/ncomms14891 (2017).

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