Comparative Analysis of Volatile Compounds in Flowers of Different Actinidia Species

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Abstract: Among the actinidia genus (Actinidia spp.) native to China, few species are grown commercially for their edible and healthy fruits. As they are dioecious plants, a lot of interest is paid to effective pollination and to insects as the most efficient pollinators. In this study we have concentrated on the composition of volatile compounds in male flowers of four different actinidia species (A. chinensis var. deliciosa, A. arguta, A. kolomikta and A. polygama) and on the diversity between male and female flower volatiles for the two most winter-hardy species (A. arguta and A. kolomikta) with growing commercial value. Analyses were provided using gas chromatography with mass spectrometry (GC-MS). In total, 120 compounds were found in 15 actinidia genotypes. However, the number of identified compounds varied between species. Different main compounds or groups of volatile compounds characterised flowers of every species. Smaller differences were observed between male and female flowers of the same species. Our results suggest that actinidia flowers could be attractive to pollinating insects.

Keywords: kiwifruit; kiwiberry; silver vine; lilac compounds; headspace samples; attractants for pollinators; pollination; Actinidia arguta; Actinidia chinensis

1. Introduction

Actinidia genus (Actinidia spp.) includes 54 species [1]. However, only few species and varieties are commercially cultivated, including Actinidia chinensis, Actinidia arguta and Actinidia kolomikta. All actinidia species are dioecious woody vines [2] and come mostly from Asia. The most popular species, cultivated worldwide, are A. chinensis var. chinensis and A. chinensis var. deliciosa (amounting to 31% and 37%, respectively, of the total actinidia production), but other species are gaining in importance. In particular, A. arguta, also known as hardy kiwifruit, kiwiberry or minikiwi, and A. kolomikta, known as super-hardy kiwi, amounting already to 17% and 8% of the total output, respectively [3]. Actinidia arguta, a smooth-skinned, grape-sized fruit, contains a high amount of vitamin C, phenolic compounds and minerals, representing a high antioxidant activity, and hence a wide range of health benefits [4–8]. Actinidia kolomikta is the most tolerant to low temperatures (up to ~40 °C). The fruit is smaller than kiwiberry and matures early in the season [2,4]. Generally, A. arguta and A. kolomikta fruits both present higher levels of antioxidants, vitamin C and concentration of minerals than A. chinensis [3]. The fruit of Actinidia polygama, which has a nice orange colour and a bitter taste, was used mainly in ancient Chinese and Korean folk medicine [9].

As actinidia are dioecious plants, a lot of attention is given to effective pollination. At first, A. arguta was thought to be a wind-pollinated plant. However, previous studies showed that insects like to visit these flowers, and this has a high impact on the fruit set [10–13]. Moreover, as was found in A. polygama, the seemingly perfect (meaning bisexual) flowers (with stamens producing sterile pollen)
Plants 2020, 9, 1675 2 of 17

are more attractive to insects than those with artificially removed stamens [14]. There are also reports that insects are sensitive to specific volatiles and that they visit certain species more on purpose rather than accidentally. Furthermore, many insect species are able to distinguish between enantiomers of different substances, but react only to one. Such behaviour was also found in the case of pollinating insects [15]. Among flower volatile compounds, lilac aldehyde is highly attractive to pollinators (e.g., Lepidoptera) and is emitted in high amounts in nocturnal plant species [16]. Twidle et al. [17,18] found that certain volatiles of both A. chinensis varieties are recognized by honey bees and bumble bees. Then, apart from the flowers’ appearance and the number of stamens, the composition of volatile compounds in flowers is also an interesting subject.

To date, more attention is paid to female rather male cultivars, and hence the offer of male selections in nurseries is very low. For various reasons, artificial pollination (often applied in the cultivation of A. chinensis) is not practised in kiwiberry production (due to the difficult access to A. chinensis and A. arguta pollen in most kiwiberry-growing countries and the potential risk of Psa transmission by imported A. chinensis pollen). Therefore, the selection of effective pollinizers, both in terms of the pollen quantity and quality as well as the attractiveness of flowers to pollinators, is needed to produce good-quality fruit. Recent research also shows that the source of pollen may have an impact on some biochemical features of kiwiberries and that a proper selection of pollinizers appears to be important in commercial fruit production [19]. Therefore, it would be worthwhile to compare the composition of volatile compounds in flowers of different species of actinidia and to ascertain the diversity within one species.

Volatile compounds in flowers of A. chinensis var. chinensis and A. chinensis var. deliciosa have been well analysed, as this species is of great economic importance. An analysis carried out on solvent and headspace samples of cut flowers showed a presence of 107 and 59 compounds, respectively. Most of them were carbonyls, alcohols, hydrocarbons and esters [20]. Another headspace analysis of cut flowers identified 27 mostly terpenoid compounds [21]. Twidle et al. [22] analysed the headspace of unpollinated flowers and identified 45 volatile compounds, mostly straight chain hydrocarbons and terpenes.

To date, little attention has been paid to volatile compounds of flowers of other actinidia species. The chemical analysis (headspace and solvent extracts GC-MS (gas chromatography—mass spectrometry)) of seven female and two male genotypes of A. arguta helped to identify some 60–70 compounds in flowers and 40–80 in fruits. The most numerous chemical groups in flowers were represented by terpenes (including lilac derivatives coming from linalool) and benzenoid compounds. A GC analysis showed differentiation in the composition of flower volatile compounds between the tested genotypes [23]. A study on a different species has shown that the flower aroma of even closely related species, such as Brunfelsia australis and B. pauciflora, may differ significantly [24]. Considering the above, understanding the composition of volatile compounds in flowers of less known species of actinidia, may help to optimize the commercial production of their fruit.

The aim of the research was to compare the composition of volatile compounds of flowers from male genotypes of A. chinensis var. deliciosa, A. arguta, A. kolomikta and A. polygama, as well as male and female genotypes of A. arguta and A. kolomikta, the two most winter-hardy actinidia species commercially cultivated in colder regions.

2. Results and Discussion

2.1. Composition of Volatile Compounds in Actinidia Flowers

Flowers of tested species differ in size. Actinidia chinensis var. deliciosa produce the biggest flowers (up to 4 cm in diameter), while A. kolomikta produce the smallest (usually ~1.5 cm in diameter). The structure of flowers appears to be less diverse. However, male and female flowers differ significantly. Male flowers contain only numerous stamens, while the female ones, seemingly perfect,
Plants 2020, 9, x 3 of 18

have—in addition to residual stamens (producing empty, non-viable pollen)—a pistil with radially arranged styles.

In total, 120 different volatiles, which can be divided into nine groups, were found. About 22 compounds were found in flowers of *A. chinensis* var. *deliciosa*, 16–17 in *A. kolomikta* (male and female, respectively), about 80 and 39 in *A. arguta* (male and female, respectively) and merely 11 compounds in *A. polygama* flowers (Figure 1). Significant differences were observed between species in the composition of volatile compounds in flowers. Hydrocarbon 9-octadecene E-dominated in *A. chinensis* var. *deliciosa*. *A. arguta* was characterised by the presence of linalool and its derivatives. Beta-myrcene was the main compound (>65%) in *A. kolomikta*, while linalool constituted almost half of the emitted volatiles in *A. polygama* (Table 1, Figure 2a–d). All identified compounds can be divided into chemical groups dominating in the particular species (Figure 1). *A. kolomikta* and *A. arguta* were characterised by a high diversity of terpenes, while only few terpenoid compounds were found in *A. chinensis* var. *deliciosa* and *A. polygama*. However, the *A. chinensis* var. *deliciosa*’s headspace sample was characterised by a significant number of straight-chain hydrocarbons. In *A. polygama* and *A. arguta* plenty of alcohols were identified, but with a difference between these two species (Table 1, Figure 2b,c). Considering all volatiles in these species, the most numerous groups were alcohols, terpenes, esters and straight hydrocarbons, amounting to about 26, 21, 13 and 12% of all the identified volatiles, respectively. Among terpenes, the largest group consisted of linalool derivatives identified in all tested *A. arguta* genotypes. On the other hand, ketones, aldehydes and acids were not typical for most of the tested species, and certain groups were observed in just one or two species.

![Figure 1](image-url)  
**Figure 1.** Differences in number and composition of volatile compounds in particular actinidia species.
Table 1. Volatile compounds identified in flowers of four actinidia species (all genotypes tested). Data represent the approximate relative percentage of peak areas for each given species. **Bold name** = mass spectra library match strength 90% or greater; **Bold Retention Indices (RI)** = positively compared to Van Den Dool and Kratz RI on https://webbook.nist.gov/.

| Compound                      | Retention Time | Retention Index * | Retention Index, Literature Data ** | A. chinensis var. delicosa | A. polygama | A. kolomikta | A. arguta Male Genotypes | A. arguta Female Genotypes |
|-------------------------------|----------------|-------------------|-------------------------------------|---------------------------|-------------|-------------|---------------------------|----------------------------|
| Terpenes                      |                |                   |                                     |                           |             |             |                           |                            |
| Geraniol                      | 28.451         | 1637              | -                                   | -                         | 0.2         | 0.3         | -                         | -                          |
| Neral                         | 29.451         | 1660              | -                                   | -                         | 0.2         | 0.4         | -                         | -                          |
| beta-Citronellol              | 31.677         | 1735              | -                                   | 13.2                      | 3.5         | 6.3         | -                         | -                          |
| D-Limonene                    | 9.114          |                   | -                                   | 0.4                       | -           | -           | -                         | -                          |
| Eucalyptol (43, 32, 81, 71, 84) | 8.812          | 1220              | -                                   | -                         | 0.2         | 0.3         | -                         | -                          |
| Alpha-Farnesene               | 31.732         | 1740              | 5.0                                 | -                         | 0.5         | -           | -                         | -                          |
| trans-Geraniol                | 34.680         | 1833              | 5.1                                 | 3.4                       | 5.1         | -           | -                         | -                          |
| 5-Hepten-2-one, 6-methyl-      | 14.895         | 1323              | -                                   | -                         | 0.9         | 2.9         | 2.6                       | 1.3                        |
| Lilac alcohol B               | 22.855         | 1462              | -                                   | -                         | -           | -           | 0.2                       | 0.4                        |
| Lilac alcohol C               | 23.199         | 1473              | -                                   | -                         | -           | -           | 0.2                       | 0.4                        |
| Lilac alcohol D               | 24.615         | 1519              | -                                   | -                         | 1.7         | 3.4         | 3.2                       | 0.4                        |
| Lilac alcohol D               | 24.947         | 1529              | -                                   | -                         | -           | 2.3         | 20.6                      | 12.7                       |
| Lilac alcohol D               | 23.987         | 1548              | 47.8                                | 0.7                       | 1.0         | 11.5        | 21.8                      | 22.8                       |
| Lilac alcohol D               | 29.696         | 1675              | -                                   | 4.0                       | 4.0         | 3.8         | 2.8                       | 0.5                        |
| Lilac alcohol D               | 30.276         | 1693              | -                                   | 1.4                       | 2.9         | 5.1         | 4.5                       | 1.3                        |
| Lilac alcohol D               | 30.967         | 1714              | -                                   | 1.0                       | 1.4         | 1.0         | 1.9                       | 0.9                        |
| Lilac alcohol D               | 32.197         | 1754              | -                                   | 1.2                       | -           | 12.0        | -                         | -                          |
| Lilac alcohol D               | 32.754         | 1772              | -                                   | 13.8                      | 3.0         | 12.5        | 10.3                      | 1.7                        |
| Lilac alcohol D               | 33.700         | 1802              | -                                   | 2.6                       | 2.3         | 12.9        | -                         | 9.5                        |
| Lilac alcohol D               | 34.228         | 1820              | -                                   | -                         | -           | -           | 9.7                       | -                          |
| Linear oxide                  | 20.602         | 1668              | 0.6                                 | -                         | 0.1         | 0.2         | 0.6                       | -                          |
| Beta-Myrcene                  | 7.649          | 1148              | -                                   | 74.7                      | 67.2        | -           | -                         | -                          |
| Cit-rutinol (41, 121, 67, 55, 29, 79) | 17.233         |                   | -                                   | 0.4                       | -           | -           | -                         | -                          |
| Nerol                         | 32.960         | 1778              | 4.0                                 | 5.8                       | -           | -           | -                         | -                          |
| alpha-oicinome                | 11.362         | 1232              | 3.9                                 | 3.6                       | 0.4         | 0.3         | 0.4                       | -                          |
| E-beta-oicinome               | 11.946         | 1230              | 1.2                                 | 3.9                       | 3.5         | 2.8         | 2.2                       | 2.1                        |
| 1,6-Octadiene, 5,5-dimethyl-trans-(69, 41, 79, 53, 82) | 13.651         |                   | -                                   | 1.7                       | -           | -           | -                         | 0.5                        |
| 5,9-Undecadien-2-one, 6,10-dimethyl- (E) | 34.002       | 1842              | 1850                                | 0.6                       | 1.1         | 2.2         | 0.7                       | 1.7                        |
| 5,9-Undecadien-2-one, 6,10-dimethyl- (Z) | 35.478       | 1857              | 1855                                | -                         | 0.4         | -           | -                         | -                          |
| 6-Octadiene, 5,5-dimethyl-trans-(69, 41, 79, 53, 82) | 19.432         |                   | -                                   | -                         | 0.2         | -           | -                         | -                          |
| Benzaldehyde                  | 22.335         | 1445              | 1504                                | -                         | 0.1         | -           | -                         | -                          |
| Benzenoid compounds           | 22.807         | 1460              | -                                   | 0.1                       | 0.4         | 0.2         | 0.3                       | 0.3                        |
| Benzenoic acid (77, 106, 105, 51, 50, 52) | 6.998          | 1140              | 0.4                                 | -                         | -           | -           | -                         | -                          |
| Benzenoic acid (77, 106, 105, 51, 50, 52) | 26.797         | 1586              | 1617                                | 0.6                       | 0.9         | 0.3         | 0.7                       | -                          |
| Compound | Retention Time | Retention Index | Retention Index, Literature Data | A. chinensis var. deliciosa | A. polygama | A. kolomikta | A. arguta Male Genotypes | A. arguta Female Genotypes |
|-----------|----------------|----------------|---------------------------------|-----------------------------|-------------|-------------|-----------------------------|-----------------------------|
| 2-(4-Methoxyphenyl) ethanol Phenylethyl Alcohol | 49.788 | - | - | - | - | - | - | 0.3 | 0.2 | 0.4 | 0.3 | 0.6 | 0.4 | - | 2.1 | 0.6 | 0.5 | 0.5 |
| Acetic acid, 1-(2-methyltetrazol-5-yl) ethyl ester | 44.771 | 2164 | - | - | - | - | - | 0.1 | - | - | - | - | - | - | - | - | - | - |
| Butanoic acid, 2-methyl-, 2-methylbutyl ester | 12.551 | - | 1274 | - | - | - | - | - | 0.3 | - | - | - | - | - | - | - | - | - |
| Butanoic acid, 3-methyl-, 3-methylbutyl ester | 13.297 | - | 1312 | - | - | - | - | 0.1 | - | 2.1 | 3.3 | 0.4 | - | - | - | 2.3 | 0.5 | - |
| Dichloroacetic acid, 4-hexadecyl ester | 26.944 | 1591 | - | 1.4 | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 5-Dodecen-1-ol, acetate, (Z)- Hexadecanoic acid, methyl ester | 46.421 | - | 2223 | - | 0.3 | 0.5 | - | - | - | - | - | - | - | - | - | - | - | - |
| Hexadecanoic acid, 15-methyl-, methyl ester | 52.424 | 2438 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | 0.2 |
| Octadecanoic acid, methyl ester | 52.532 | 2417 | - | 0.2 | 0.7 | - | - | - | 1.6 | - | - | - | - | - | - | - | - | - |
| 9-Octadecenoic acid (Z), methyl ester | 46.027 | 2207 | 2430 | - | 0.5 | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 10-Octadecenoic acid, methyl ester | 52.928 | 2455 | - | 0.9 | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 12,15-Octadecadienoic acid, methyl ester | 54.272 | 2503 | - | 0.2 | - | 0.3 | - | - | - | - | - | - | - | - | - | - | - | - |
| 2-Phenylethyl acetate | 33.237 | 1757 | 1777 | 0.4 | - | - | - | - | - | - | - | - | - | 0.1 | 0.6 | - | - | - |
| Propanoic acid, 2-methyl-, 3-hydroxy-2,4,4-trimethylpentyl ester | 35.558 | 1860 | - | - | - | - | - | 0.2 | 0.3 | - | - | - | - | - | - | - | - | - |
| 2-Propenoic acid, 3-phenyl-, pentyl ester | 66.841 | - | - | - | - | - | - | 0.2 | - | - | - | - | - | - | - | - | - | - |
| 11-Tetradecen-1-ol, acetate, (Z)- | 6.825 | 2137 | - | - | - | - | 3.4 | - | - | - | - | - | - | - | - | - | - | - |
| Aldehydes | | | | | | | | | | | | | | | | | | | |
| Furfural (39, 96, 95, 79, 29, 38) | 20.040 | - | 1432 | - | 1.3 | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 2-Hexenal | 9.667 | - | 1225 | - | 6.6 | - | - | - | - | - | - | - | - | - | - | - | - | - |
| Nonanal | 21.166 | 1403 | 1395 | - | - | - | - | - | 0.1 | - | - | - | - | - | - | - | - | - |
| 5-methyl furfural (110, 53, 109, 27, 39, 41) | 24.344 | 1511 | 1560 | - | 0.3 | - | - | - | - | - | - | - | - | - | - | - | - | - |
| Ketones | | | | | | | | | | | | | | | | | | | |
| Cyclopentanone, 2-methyl- (45, 55, 28, 43, 42, 41) | 12.529 | - | - | - | - | - | - | - | - | 0.2 | - | - | - | - | - | - | - | - |
| 3,4-Dimethyl-2-pentanone (43, 55, 71, 28, 114, 41) | 43.604 | 2124 | - | - | - | - | 0.1 | - | - | - | - | - | - | - | - | - | - | - |
| 2,6-Dimethyl-6-nitro-2-heptan-4-one (83, 55, 43, 29, 27, 39) | 32.005 | 1748 | - | 0.2 | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 4-Heptanone, 3-methyl- | 26.480 | 1576 | - | - | - | 0.1 | - | - | - | - | - | - | - | - | - | - | - | - |
| 5-Heptanone-3-one, 4-methyl- | 14.312 | - | 1395 | - | - | - | - | - | 0.3 | - | - | - | - | - | - | - | - | - |
| 3-Octanone | 11.460 | - | 1205 | - | - | - | - | - | 4.2 | - | - | - | - | - | - | - | - | - |
| 2-Pentanone, 4-hydroxy-4-methyl- (43, 59, 56, 42, 41, 207) | 15.961 | - | 1352 | - | - | - | - | - | 0.2 | - | - | - | - | - | - | - | - | - |
| 4H-Pyran-4-one, 2,3-dihydro-3,5-dihydroxy-6-methyl- | 47.988 | 2278 | 2274 | - | - | - | - | - | - | 0.2 | - | - | - | - | - | - | - | - | - |
### Table 1. Cont.

| Compound                  | Retention Time | Retention Index * | Retention Index, Literature Data ** | A. chinensis var. deliciosa | A. polygama | A. kolomikta | A. arguta Male Genotypes | A. arguta Female Genotypes |
|---------------------------|----------------|-------------------|-------------------------------------|-----------------------------|-------------|-------------|--------------------------|-----------------------------|
| **Alcohols**              |                |                   |                                     |                             |             |             |                          |                             |
| 1,4-Butanediol            | 37.292         | 1915              |                                     | Male                        | Male        | Male        | 'Nostino'                | 'Joker'                     |
| Butanol, 2-methyl-         | 9.887          | -                 | 1201                                | Male                        | Male        | Male        | 12.1                      | 1.5                         |
| 1-Butanol, 3-methyl- (impure) | 9.577      | -                 | 1210                                | Female                      | Female      | 'Rot'       | 'Weiki'                  | 'Rubi'                      |
| 2-Buten-1-ol, 3-methyl- (71, 41, 29, 39, 27) | 14.530 | -                 | 1301                                | 0.2                         | -           | 0.2         | 38.5                      | 49.4                        |
| 1-Decanol                 | 32.153         | 1753              | 1760                                | 0.5                         | -           | -           | 48.4                      | 36.3                        |
| 1,1-Dimethyl-3-chloropropanol (59, 43, 41, 31, 27, 28) | 12.889 | -                 |                                     | 0.2                         | -           | -           | 12.1                      | 4.4                         |
| Ethanol, 2-(2-ethoxyethoxy) (45, 31, 59, 29, 72, 27) | 26.644 | 1382              | 1615                                | -                           | -           | -           | 12.1                      | 42.9                        |
| 2-Hepten-3-ol, 4,5-dimethyl- (71, 43, 32, 29, 27) | 20.468 | -                 | -                                   | -                           | -           | -           | 36.3                      | 26.9                        |
| 6-Hepten-1-ol, 3-methyl- (55, 71, 73, 43, 67, 41) | 54.185 | 2501              | 2901                                | 0.6                         | -           | -           | 0.2                       | -                           |
| 1-Hexanol                 | 15.314         | -                 | 1356                                | 0.2                         | 0.3         | 0.3         | 17.5                      | 6                          |
| 1-Hexanol, 3-methyl- (56, 55, 72, 60, 26) | 24.406 | 1515              | 1413                                | -                           | -           | -           | 0.2                       | -                           |
| 2-Hexen-1-ol, (E)         | 17.570         | -                 | 1403                                | -                           | -           | -           | 0.2                       | -                           |
| 2-Hexen-1-ol, (Z)         | 17.948         | -                 | 1380                                | 0.6                         | -           | -           | 0.2                       | -                           |
| 3-Hexen-1-ol, (Z) (41, 67, 55, 27, 29, 82) | 15.715 | -                 | 1373                                | 0.6                         | -           | -           | 36.3                      | 1.1                        |
| 3-Methyl-hepta-1,6-dien-3-ol (71, 43, 28, 55, 57, 29) | 20.409 | -                 | 1356                                | 0.3                         | 0.3         | 0.3         | 0.7                       | -                           |
| 4-Methyl-1-heptyn-3-ol (43, 71, 35, 27, 82, 41) | 43.636 | 2126              | -                                   | 0.1                         | -           | -           | -                         | -                           |
| 9,12-Octadecadien-1-ol (67, 81, 55, 52, 96) (5Z)-Octa-1,5-dien-3-ol (57, 70, 41, 29, 55, 27) | 36.871 | 1901              | -                                   | 0.2                         | 0.2         | 0.2         | 0.2                       | -                           |
| 1,7-Octadecen-3-ol, 2,6-dimethyl- (71, 43, 55, 27, 39) | 44.471 | 2154              | -                                   | 0.8                         | 0.8         | 1.1         | 1.6                       | 0.8                         |
| 1,5,7-Octatrien-3-ol, 3,7-dimethyl- (71, 43, 55, 27, 39) | 25.870 | 1558              | 1613                                | 1.4                         | 1.6         | 0.8         | 16                        | 1.4                         |
| Benzoic acid              | 52.562         | 2443              | 2444                                | 0.2                         | 0.3         | 0.2         | 0.2                       | -                           |
| n-Hexadecanoic acid       | 65.080         | -                 | 2880                                | 0.2                         | 0.3         | 0.2         | 0.3                       | -                           |
| 2-Methylbutanoic acid     | 28.485         | 1638              | 1641                                | -                           | -           | -           | -                         | -                           |
| 9,12,15-Octadecatrieniox acid, (ZZZ)-(79, 55, 41, 67, 93, 95) | 34.191 | 1818              | 3292                                | 0.2                         | 0.2         | 0.2         | 0.2                       | -                           |
| Straight hydrocarbons     | 24.713         | 1532              | 0.2                                 | -                           | -           | -           | -                         | -                           |

* Retention Index: Higher values indicate a longer retention time.

** Retention Index, Literature Data: Values from the literature indicate the retention index from different sources, which may vary.

Data Source: Literature **
| Compound                        | Retention Time | Retention Index | Retention Index, Literature Data | A. chinensis var. delicosa | A. polygama | A. kolomikta | A. arguta Male Genotypes | A. arguta Female Genotypes |
|--------------------------------|----------------|----------------|----------------------------------|---------------------------|-------------|--------------|--------------------------|---------------------------|
| 9-Eicosene, (E)-               | 37.632         | 1926           | -                                | 1.4                       | -           | -            | -                        | -                         |
| Heptadecane                    | 30.150         | 1689           | 1700                             | 3.0                       | -           | -            | -                        | -                         |
| 8-Heptadecane                  | 29.916         | 1682           | 1718                             | -                         | -           | -            | -                        | -                         |
| 5-Heptadecane, 1-bromo-        | 35.472         | 1857           | -                                | 0.7                       | -           | -            | -                        | -                         |
| Heptadecane, 1-fluoro-         | 21.693         | 1424           | -                                | -                         | -           | -            | -                        | -                         |
| Hexadecane                     | 26.277         | 1570           | 1600                             | 1.77                      | -           | -            | -                        | -                         |
| 2-Hexene, 3,5,5-trimethyl-     | 21.405         | 1415           | -                                | -                         | -           | -            | -                        | -                         |
| Nonadecane                     | 37.199         | 1912           | 1900                             | 0.7                       | -           | -            | -                        | -                         |
| 9-Octadecene (E)-             | 30.799         | 1708           | -                                | 31.8                      | -           | -            | -                        | -                         |
| Pentadecane                    | 24.661         | 1637           | -                                | 0.2                       | -           | -            | -                        | -                         |
| 5-Tetradecene, (E)-           | 23.875         | 1495           | 1500                             | 19.6                      | -           | -            | -                        | -                         |
| Tridecane                      | 19.268         | -              | 0.35                             | -                         | -           | -            | -                        | -                         |
| Others                         |                |                |                                  |                           |             |              |                          |                           |
| Butyrolactone                  | 25.260         | 1612           | 1617                             | -                         | -           | -            | 0.2                      | 0.3                       |
| gamma-Butyrolactone            | 26.124         | 1567           | -                                | -                         | 0.3         | 0.5          | 0.5                      | 1.1                       |
| Dibydroxyacetone               | 42.282         | 2080           | -                                | -                         | -           | -            | -                        | -                         |
| 2,5-divinyl-2-methyltetrahydrofuran | 7.724   | -              | 0.5                              | -                         | 0.2         | 2.4          | 0.3                      | 0.3                       |
| Furan, 3-(4-methyl-3-pentenyl)- | 18.103       | 1431           | -                                | 4.2                       | -           | -            | 0.7                      | -                         |
| 2-Furancarboxaldehyde          | 54.536         | 2513           | 2513                             | -                         | -           | -            | -                        | -                         |
| Gearyl bromide                 | 13.693         | -              | 0.5                              | -                         | -           | -            | -                        | -                         |
| 1,4-Hexadiene, 5-methyl-3(1-methylene) | 16.338    | -              | 3.7                              | 3.6                       | 1.5         | 1.5          | 1.4                      | 0.5                       |
| Hexanoyl chloride              | 11.421         | -              | 0.7                              | -                         | -           | -            | -                        | -                         |
| 1-Mimidazol-1-yl-2,2-dimethylpropan-1-one | 17.392 | -              | 0.3                             | -                         | -           | 0.3          | -                        | -                         |
| Isoamyl cinnamate              | 66.838         | -              | 0.4                              | -                         | -           | -            | -                        | -                         |
| 1,3-Pentadiene, 5-(2,2-dimethylcyclopropyl)-2,4-dimethyl, (Z or E)- | 17.358 | -              | -                                | 0.3                       | -           | -            | -                        | -                         |

* Retention Indices were calculated according to van Den Dool and Kratz (1963); ** Retention Indices according to https://webbook.nist.gov/, Van Den Dool and Kratz RI for polar column and, if possible, temperature ramp identical or at least similar to this analysis; *** Mass spectral fragments for unconfirmed compounds in descending order.
were found by Tatsuka et al. [20]. Among the 27 compounds identified by Samadi-Maybodi, Shariat and var. A. chinensis (0.7%)—were also found in the (phenylethyl alcohol) was also found in (0.03–0.13%) [22]. As in previous research, a similar level of pentadecane was found in (E) and pentadecane, were identified as the main compounds and accounted for 31.8 and 19.6%,

2.2. Differences in the Composition of Volatile Compounds in Flowers of Different Actinidia Species

In our research, much of the A. chinensis var. deliciosa headspace samples were straight-chain hydrocarbons. None of these were found in other tested species (Table 1). Two of these, 9-octadecene (E) and pentadecane, were identified as the main compounds and accounted for 31.8 and 19.6%, respectively (Figure 2a), of the total. These two compounds were previously identified in A. chinensis. 9-octadecene (E) was found in A. chinensis var. deliciosa cultivars, yet in a much smaller percentage (0.03–0.13%) [22]. As in previous research, a similar level of pentadecane was found in A. chinensis var. chinensis and in A. chinensis var. deliciosa (9–20%, depending on the cultivar) [20,22]. 5-dodecen-1-ol, acetate, (Z)- (19.9%) and phenylethyl alcohol (9.2%) represented other important compounds. The latter (phenylethyl alcohol) was also found in A. arguta male (0.2–12.2%) and female (5.3–25.0%) flowers. A few more aromatic straight-chain hydrocarbons—heptadecane (3.0%), hexadecane (1.8%), and nonadecane (0.7%)—were also found in the A. chinensis headspace. Finally, 2-phenethyl acetate (0.4%), and benzoic compound—1,3-dimethylbenzene (0.4%) were identified in the A. chinensis var. deliciosa sample. A. chinensis var. deliciosa headspace was characterised by the presence of a few terpenes. These were: alpha-Farnesene, ocimene, cis-Geranylacetone and D-limonene (the one with a well-known lemon-like odour). Their content in headspace amounted to 5.0, 1.2, 0.4 and 0.4%, respectively. D-limonene was also found previously in A. chinensis flowers [20,21].

Comparison to previous research on both A. chinensis varieties indicated that the particular genotypes differed in the number of identified compounds, composition and amount [17,18,20–22]. Five of the compounds identified by Twiddle et al. [18] in A. chinensis var. deliciosa, namely alpha-Farnesene, hexadecane, cis-geranylacetone, nonadecane and phenylethyl alcohol, were also found in our research. Similar compounds such as those in our research, namely terpenes (D-limonene, ocimene, alpha-Farnesene), straight-chain hydrocarbons (pentadecane) and alcohols, like phenylethyl alcohol, were found by Tatsuka et al. [20]. Among the 27 compounds identified by Samadi-Maybodi, Shariat and Zarei [21] in A. chinensis, only three (limonene, ocimene and alpha-Farnesene) were identical to the ones in our research. The following compounds were found in our and previous research in A. chinensis

Figure 2. Sample chromatograms with main volatile compounds in male flowers of four actinidia species: (a) Actinidia chinensis var. deliciosa; (b) Actinidia polygama; (c) Actinidia kolomikta; (d) Actinidia arguta.
var. *deliciosa* flowers: tridcane, nonadecane, pentadecane, hexadecane, 7-hexadecene, heptadecane, alpha-Farnesene, 2-phenylethyl acetate, phenylethyl alcohol and cis-geranylacetone. However, the latter compound was found in one of the six genotypes of *A. chinensis* var. *deliciosa* [22]. Alpha-Farnesene was the only compound found in all available results; however, its percentage of all the identified compounds stood between 0.6 and 38.0%, depending on the information source [17,18,20–22]. No linalool and lilac compounds characteristic of *A. arguta* were found in *A. chinensis* [23,25].

When analysing the *A. polygama* flower aroma in detail, it was found that some compounds were characteristic of that species only. These included, e.g., 2-hexenal (6.6%) and 2-hexen-1-ol, (E)- (17.5%) (a primary allylic alcohol, used for food flavouring, PubChem). Form (Z)- was also found, but in a much smaller amount (0.2%). However, linalool (47.8%) (Figure 2b), the major compound of *A. polygama* volatiles, was also discovered in eight out of eleven *A. arguta* samples. Beta-citronellol (13.2%) was another important compound of *A. polygama* also found in *A. kolomikta*, yet in a minor amount (3.5%) (Figure 2b,c). Generally, only few volatiles of *A. polygama* were found in other species. Along with the already mentioned linalool and beta-citronellol, 3-hexen-1-ol, (Z) (4.5%) and 1- hexanol (8.6%) were identified. Both were also found in two male *A. arguta* genotypes (‘Rot’, ‘Rubi’). Additionally, 1-hexanol was present in ‘Nostino’ and ‘Joker’. It is worth noting that no common volatile compounds for *A. chinensis* and *A. polygama* flowers were identified.

In *A. kolomikta*, monoterpane β-myrcene was the compound representing the highest relative content (~75% in male and ~67% in female) (Table 1, Figure 2c). Myrcene, an isomer of ocimene, is said to be unstable in the air and tends to polimerate easily [26] (p. 280). Nerol and geraniol (4.0% and 3.4%, respectively), identified in male *A. kolomikta*, turned out to be characteristic of this species alone and were not traced in other actinidia species. Another characteristic feature of *A. kolomikta* compounds were citral isomers (geranial and neral), responsible for the lemon note in the aroma. Neral accounted for less than 1%, yet its aroma is very strong. Certain volatiles were found not only in *A. kolomikta*, but also in other actinidia species. Citronellol was one of them, mentioned above, while the others were terpens ocimene α and β, found in *A. kolomikta* and *A. arguta*. Beta-ocimene was also identified in *A. chinensis* var. *deliciosa*’s flowers. Other substances found in *A. kolomikta* and in a few *A. arguta* genotypes were alpha-Farnesene, 1,4-Hexadiene, 5-methyl-3-(1-methylethylidene) and a group of esters, with each of these accounting for less than 1%. Alpha-Farnesene was also found in *A. chinensis* var. *deliciosa*.

Summing up the above, it appears that the number of volatile compounds and their composition differ in the particular actinidia species. Moreover, every species seems to have one or few dominating compounds affecting their flower aroma.

2.3. Diversity of Volatile Compounds in Flowers of Different *A. arguta* Genotypes

*Actinidia arguta* is the species with the largest area of natural occurrence, both in terms of latitude and altitude [27]. This affects the large morphological diversity within the species (e.g., in terms of leaf shape, size, shape and colour of the fruit), which may account for the richest composition of flower volatile compounds, belonging to at least eight different chemical groups. Among the 120 detected compounds, about 80 were found in seven male clones of *A. arguta* (Table 1, Figure 3). Of these, ten (lilac aldehydes, lilac alcohols, 5-Hepten-2-one, 6-methyl-, geranyl acetone, phenylethyl alcohol, 1,7-Octadien-3-ol, 2,6-dimethyl-, 1,4-Hexadiene and 5-methyl-3-(1-methylethylidene)) were detected in all seven male clones. Their relative content varied depending on the genotype, even in the same sex (Figure 3). However, about 44 compounds (55%) were detected only in one genotype, and nine in two. These were mainly esters, ketones, alcohols and straight-chain hydrocarbons. Few terpenoid volatiles, excluding linalool and ocimene derivatives, were found only in one or two genotypes. The relative content of these compounds was usually small and accounted for less than 1%.
In this research, linalool derivatives represented the most important group found in *A. arguta* flowers. Linalool, lilac aldehydes and alcohols were present in almost all samples (Figure 2d). Differences in their composition in headspace samples were high, and their summed content varied between 24.3% and 92.7%. A detailed analysis showed that in few instances more than one peak was identified as representing the same compound (among lilac aldehydes and alcohols). That was probably due to not distinguishing isomers by mass spectrometer. The analysis of retention indices was not conclusive either, as only partial information on the compounds analysed is available in the literature and online data. An additional detailed analysis is needed to clarify whether these peaks are indeed two isomers of the same compounds. When summing up all lilac aldehydes for each genotype, it follows that differences were high. The lowest percentage of the compound was found in ‘Haya Kume’ (2.7%), while the headspace of the other clones contained at least 17.0% (‘Weiki’). In the *A. arguta* ‘Rot’, which was previously classified as *A. arguta var. purpurea* (form with red fruit), the content of lilac aldehyde amounted to 25.0%. According to the latest genetic revision [1], *A. arguta var. purpurea* has been included in the *A. arguta* variability. The highest content was observed in flowers of clone F7 (53.0%), which is a hybrid between *A. arguta* and *A. arguta var. purpurea*. This may suggest, however, that the offspring of this cross might produce a higher amount of lilac aldehyde in flowers. The analysis of lilac alcohol led to a similar observation. The highest percentage of the compound was observed for F7 (39.7%) and the lowest for ‘Haya Kume’ (5.4%). *Actinidia arguta* ‘Rot’ represented a rather low content of lilac alcohol in flowers (12.9%). The sum of all isomers of lilac aldehyde and alcohol ranged between 38 and 53% for most of the tested genotypes. The lowest content of these compounds in the headspace has been ascertained in the ‘Haya Kume’ sample (8.1%), which also had the least intense aroma, organoleptically (own observation, unpublished). The content of lilac alcohol and aldehyde isomers exceeded 92% in the F7 genotype. It is noteworthy that in two male
samples, ‘Weiki’ and F7, no linalool was detected. Isomers of linalool derivatives in A. arguta were identified by Dötterl et al. [16]. The authors found two stereoisomers of linalool (S, R), four of lilac aldehydes and four of lilac alcohol. The same combination of stereoisomers was found in two out of 15 species investigated (Cynanchum auriculatum and Viburnum opulus). Some species contained one stereoisomer of linalool and the same (Syringa vulgaris) or similar (e.g., Silene oites, Prunus padus and Daphne cneorum) combination of lilac alcohol and aldehyde stereoisomers. Further analysis revealed that derivatives of linalool were more complex than expected. In earlier research of A. arguta, these derivatives were present only in a few examined genotypes, and some of these were found in solvent extracts [23,25]. When investigating the stereoisomers of lilac aldehyde, it appears that our results differ from those by Matich et al. [23], where only one headspace male flower sample missed one lilac aldehyde stereoisomer, whereas in this particular study only lilac aldehydes C and D were found. However, as observed earlier, the problem could be due to an inappropriate distinction of lilac stereoisomers. In a previous study, four stereoisomers of lilac alcohol were also found in headspace samples [23], while only two forms were found in our analysis. A comparison to the insightful research undertaken by Matich et al. [23] shows that lilac compounds, which were first isolated from Syringa vulgaris flower oil and detected in at least nine plant families [16], are important components of A. arguta headspace samples. In our analysis, however, the relative content of both lilac alcohol and aldehyde was higher than in Matich et al.’s [23] analysis. The reason may be methodological (that is, the use of a different method of collecting samples) as well as genetical. 5-Hepten-2-one and 6-methyl were also detected by Matich et al. [23] in all headspace and solvent samples, but the content amounted to less than 1% in all genotypes.

Ocimene derivatives constitute another important group among terpenes. In this work, E-beta ocimene was identified in most samples, and its relatively highest content (7.0%) was found in ‘Haya Kume’. Geranylacetone (found, e.g., in Nelumbo nucifera oil) was present in all clones and 5-Hepten-2-one, 6-methyl- (a component of volatile oils of citronella, lemon-grass and palmarosa oil) in most (e.g., 4.9% for ‘Rubi’). 5-Hepten-2-one and 6-methyl were also detected by Matich et al. [23] in all headspace and solvent samples, but the content amounted to less than 1% in all genotypes. Other terpenes identified in this analysis were previously found only in single samples. However, alpha-Farnesene, detected here in the headspace, was earlier found only in the solvent extract of one female genotype.

Only few benzoid compounds were identified in the headspace samples. One was the already mentioned phenylethyl alcohol. The highest content was noticed in ‘Joker’ (12.2%), but it stood below 5% in most of the clones. Two benzoid compounds (e.g., 2-(4-Methoxyphenyl) ethanol) were found in six out of seven male clones, but their content was low (< 0.5%). Benzeneacetaldehyde, which is an aromatic compound, was identified in four clones, but its content was low as well (< 1%). The results of Matich et al. [23] suggest that benzoid compounds are not common in the headspace, but high amounts are present in solvent samples (up to 30.1%). The only compound, identified by them in larger amounts in volatile samples, was phenylethyl alcohol (6.25 and 17.2%). That was confirmed in our research. 2-(4-Methoxyphenyl) ethanol was also identified by Matich et al. [23], yet only in solvent extracts. It was found there in small amounts but in almost all samples. Nor was benzeneacetaldehyde found in previous studies, but benzaldehyde and benzene (the compounds that were not found in our genotypes) were detected in several samples. This suggests that among benzoid compounds only phenylethyl alcohol is specific to A. arguta, in male and female genotypes alike. The identification of other compounds may depend on their concentration in the headspace, detection possibilities, genotype or environmental impact.

More than ten ester compounds were identified in all samples of A. arguta. Of these, butanoic acid, 3-methyl- and 3-methylbutyl ester, a constituent of banana fruit, tomatoes and alcohol beverages (e.g., cognac or cider), was found to be the compound that may have an impact on the aroma. The ester was found in four male samples. The content in two (‘Nostino’ and ‘Haya Kume’) amounted to less than 0.5%, and in other two (‘Rot’ and ‘Weiki’) to 2.1 and 3.3%, respectively. More esters were found in
this research than in previous ones in general. Reports of using SMPE sampling for the analysis of fatty acid esters, whose particles are heavier than those identified in our research [28], suggest that the sampling methods used contributed to a better ester detection in this research.

Only three aldehydes (furfural, 5-methylfurfural and nonanal) were detected in this research. Furfural and 5-methylfurfural were present in small amounts in the ‘Nostino’ sample (1.3% and 0.3%, respectively) and nonanal in ‘Joker’ (0.1%), whereas Matich et al. [23] identified 11 different aldehydes. Four, including nonanal, were present in all samples. Ketones were detected in two male genotypes: ‘Nostino’ and ‘Haya Kume’, but the amounts of particular compounds were different. Most of them also amounted to less than 1% of the total headspace, but the contents of one ketone—3-Octanone, detected in ‘Haya Kume’—were higher and reached 4.2%. The analysis of ketones in both studies shows that genotypes tend to have individual combinations of these compounds or almost no ketones.

High differences in relative content were also noticed for two alcohols: 1-Butanol, 2-methyl- and 1-Butanol, 3-methyl. Genotypes with a high content of lilac compounds (F7) presented a low content of these alcohols (1.5% in F7 headspace); and, conversely, where a low content of lilac compounds was found, a high percentage of these alcohols was observed (up to 49.4% in ‘Haya Kume’). These two alcohols were also found by Matich et al. [23] in the headspace of A. arguta samples. In our and Matich et al.’s [23] research, one or the other compound was present in the sample, but never both of them simultaneously. 1-Octen-3-ol was also found in previous research, but only in two genotypes. Generally, few alcohols detected in our research were also found by other researchers, but a few different compounds were also identified. Some were chemically similar, e.g., in terms of the same base, but different in the location of a double chemical bond or substituent. This usually has a significant impact on a compound’s characteristics. Straight-chain hydrocarbons, like acids, were rather uncharacteristic for A. arguta flower volatiles. As in our analysis, Matich et al. [23] detected only few hydrocarbons, and most were found in just one genotype. In this work, only a few acid compounds were found, and they do not seem to have an impact on the aroma of A. arguta flowers.

Summing up, lilac compounds are the most important compounds in most A. arguta male genotypes. Alcohols also play an important part in the aroma, but a number of these substances were found only in a few or just one genotype. Some alcohols represent regular components of the aroma but differ in their relative content. Moreover, every genotype/variety has its own unique volatiles composition. This offers the possibility of selecting genotypes with a higher content of volatile compounds, and enhances their attractiveness for pollinating insects for actinidia commercial orchards.

2.4. Male Versus Female A. arguta and A. kolomikta Genotypes Comparison

Actinidia arguta’s female genotypes were characterised by a composition of volatiles similar to that of male ones (Figure 4a,b). However, less compounds (39) were detected in female ones. The reason could be due to the smaller number of genotypes, considering that most compounds in male samples were detected in only one genotype. Among female genotypes, important compounds were also represented by lilac compounds (Table 1). Their overall content varied between 29.5% for ‘Geneva’ and 85.4% for the ‘Bingo’ cultivar. The latter figure was much higher than the one observed in other varieties. As ‘Bingo’ is a crossing of A. arguta male and A. arguta with red fruit (previously classified as A. arguta var. purpurea), our previous observation of male genotypes was confirmed to the effect that such combination represents a higher content of linalool derivatives. In our analysis, as in that of Matich et al. [23], a higher percentage of linalool was observed in male genotypes. A relatively higher content of phenylethyl alcohol was noted in female than in male genotypes.

Interestingly, different esters were identified in flowers of male and female genotypes (Figure 4c,d). Only one ester (butanoic acid, 3-methyl-, 3-methylbutyl ester) was detected in both sexes, but still not in every genotype. No aldehydes, ketones or acids were found in female A. arguta genotypes. The composition of alcohols was smaller in female flowers, but high amounts of 1-Butanol and 3-methyl (main alcohol of male flowers) were also found in female genotypes.
19 compounds attractive to the investigated pollinators were found in our A. chinensis and B. lapidarius.

The differences between sexes in A. kolomikta were observed in the relative content of the main compound, beta-myrcene (e.g., 74.7% in the male and 67.2% in the female clone), and in a few other aromatic terpenes, like beta-citronellol, trans-geraniol and nerol (Table 1).

Summing up, the female flowers of A. arguta may have a simpler composition of volatiles than the male ones, but the observation of A. kolomikta suggests that two randomly chosen genotypes of this species may be very similar to each other.

2.5. Pollinators’ Interest

Since actinidia is dioecious, effective pollination is essential to obtain well-formed fruit. Honey bees seem to be the most important pollinators for kiwifruit [29]. Bumble bees are also used for pollination, especially since they are commercially available [22]. Our unpublished observation of kiwiberry orchards in Polish conditions suggests that A. arguta flowers are visited mainly by Bombus terrestris, B. lapidarius and Apis mellifera. According to Twidle et al. [18], honey bees and bumble bees are interested in certain A. chinensis volatile compounds from male and female flowers. Some of these compounds also found in our research, e.g., phenylethyl alcohol, alpha-Farnesene, linalool, 6-methyl-5-hepten-2-one, geranyl acetone, 2-phenylethyl acetate and 8-heptadecene, were the same or similar isomers found in at least one sample of A. arguta flowers in our research. Phenylethyl alcohol and 6-methyl-5-hepten-2-one were present in most samples. The first compound, as Twidle et al. [17,18] found out, provided the largest response from antennae in both honey and bumble bees. Furthermore, seven out of 19 compounds attractive to the investigated pollinators were found in our A. chinensis var. deliciosa sample. Honey bees were observed to respond to phenylethyl alcohol, 1-hexanol and (Z)-3-hexen-1-ol. These compounds were also attractive to nocturnal bees—major pollinators for cambuci plants.
(Campomanesia phaea) [30]. In other reports, linalool was observed to attract nocturnal bees Megalopta on guarana (Paullinia cupana) [31], and honey bees on Medicago sativa [32] and pears (Pyrus communis), where (E)-β-ocimene and lilac alcohols were also found to be attractive to bees [33]. Compounds like α-pinene and D-limonene were also found to be attractive to A. mellifera [34], with the second one identified in the A. chinensis’ headspace sample.

A comparison of our data on volatile compounds in actinidia flowers and the literature about insect interest allows us to conclude that actinidia flowers may be attractive to pollinating insects, which makes it possible to select the best genotypes, especially male ones.

3. Materials and Methods

3.1. Plant Material

The experiment was carried out in 2019. Mature plants of each genotype were grown in the Experimental Garden of the Environmental Protection and Dendrology Department at the Warsaw University of Life Sciences, in central Poland. Flowers of the following genotypes were analysed: male—A. chinensis var. deliciosa, A. polygama, A. kolomikta and A. arguta; female—A. kolomikta and A. arguta. The last species was represented by four female and seven male genotypes (Table 2).

### Table 2. Plants material used in the experiment.

| Species                  | Gender | Genotype  | Place of Plants Origin                  |
|--------------------------|--------|-----------|-----------------------------------------|
| A. chinensis var. deliciosa | male   | seedling  | Lublin Botanical Garden, Poland         |
| A. polygama              | male   | seedling  | USDA Germplasm Repository, Corvallis, US |
| A. kolomikta             | male   | ‘Adam’    | “Clematis the Source of Good Climbers”  |
| A. kolomikta             | female | ‘Tallinn’ | Private collection, Poland              |
| A. arguta                | male   | ‘Hay Kume’| USDA Germplasm Repository, Corvallis, US |
| A. arguta                | male   | ‘Joker’   | Own selection                           |
| A. arguta                | male   | ‘Nostino’ | Haeberti Nursery, Switzerland           |
| A. arguta                | male   | ‘Rot’     | Werner Merkel, Germany                   |
| A. arguta                | male   | ‘Rubí’    | Own selection                           |
| A. arguta                | male   | ‘Weiki’   | “Clematis the Source of Good Climbers”  |
| A. arguta                | male   | F7        | Own selection                           |
| A. arguta                | female | ‘Ananasnaya’(= ‘Anna’) | Fachhochschule Weihenstephan, Freising, Germany |
| A. arguta                | female | ‘Bingo’   | Own selection                           |
| A. arguta                | female | ‘Geneva’  | “Clematis the Source of Good Climbers”  |
| A. arguta                | female | ‘Weiki’   | “Clematis the Source of Good Climbers”  |

3.2. Identification and Determination of Volatile Compounds

3.2.1. Collecting Flowers and Volatile Samples

Twenty-five flowers of each genotype were collected from just opened flowers in the middle of the full flowering period. They were then put in 250 mL glass bottles, transported immediately to the lab and kept in ambient temperature for an hour before analysis. An exception was A. chinensis var. deliciosa, where only five flowers were collected due to the flower size.

3.2.2. Determination of Volatile Compounds

GC-MS was used for the identification and determination of volatile compounds. The GC-MS analyses were carried out using a mass spectrometer coupled with a gas chromatograph (GCMS-QP2010, Shimadzu, Japan) fitted with a Stabilwax column (crossbond carbowax polyethylene glycol phase) (30 m; 0.25 mm i.d.; 0.25 mm film thickness). The procedure was based on Bertrand, Comte and Piola [24], developed for Brunfelsia sp. flowers, with some modifications. The oven temperature was programmed from 50 °C to 250 °C at 3 °C/min; the injector temperature was 250 °C; the carrier gas, helium, was adjusted to a constant flow of 1 mL/min; injection type: splitless. EI-MS: electron energy, 70 eV; ion source temperature and connection parts: 250 °C. Headspace volatiles were collected using the SPME (DVB/CAR/PDMS) fibre (Sigma-Aldrich, Darmstadt, Germany). The fibre was
put into the bottle for 20 min at room temperature and for the following 2 min in 250 °C in a GC injector. The identification of volatile compounds was made on the basis of mass spectral libraries of the National Institute of Standards and Technology (NIST 47, NIST 147) and Wiley 175, as well as data from the literature [23]. Retention indices were calculated according to van den Dool and Kratz [35], dedicated to a linear temperature programmed GC. The calculations were based on the retention times of saturated alkanes C7–C30. Retention indices were calculated for the compounds with retention time ranging between 20.965 and 56.909 min, and compared to data available online at https://webbook.nist.gov/. The peak shapes of saturated alkanes below 20.965 and after 56.909 min were not satisfactory as they lacked a clear central value, and prevented the accurate determination of their retention times. In consequence, the calculation of the retention indices of volatile compounds was not possible, and additionally, retention times for all compounds were added to the Table 1. If the similarity to mass spectra libraries was below 90% and the comparison of retention indices was not possible, the most abundant six mass spectral ion fragments were also listed in the table in descending order. The peak shapes of saturated alkanes below 20.965 and after 56.909 min were not satisfactory as they lacked a clear central value, and prevented the accurate determination of their retention times. In consequence, the calculation of the retention indices of volatile compounds was not possible, and additionally, retention times for all compounds were added to the Table 1. If the similarity to mass spectra libraries was below 90% and the comparison of retention indices was not possible, the most abundant six mass spectral ion fragments were also listed in the table in descending order. For important compounds, ten fragments were reported. The results were rounded to one decimal place and presented as relative percentages of peak areas of the total ion count in the samples, separately for each genotype.

4. Conclusions

Summing up the above, 120 volatile compounds were found in total. The biggest group were terpenes and alcohols—31 and 25 compounds, respectively. The highest number of volatiles were found in A. arguta genotypes, about 80 compounds were found in seven male genotypes (but 44 compounds were found only in a single genotype, and 10 in all), and about 39 in four female genotypes. In other species, less volatiles were found: male A. chinensis var. deliciosa—22, A. kolomikta—17 in the female and 16 in the male genotype, and A. polygama male—11 compounds. A high diversity of volatile compounds was observed between species, e.g., A. polygama and A. chinensis var. deliciosa had no common substances. In all species, a characteristic compound or group of compounds could be found. However, differences within species were not so clear, e.g., the composition of volatiles in A. kolomikta was very similar in both sexes, but volatiles of A. arguta genotypes were more diverse, and this was more related to the plants’ origin than their sex.

Considerable differences in the composition of volatile compounds in male A. arguta genotypes indicate their possible selection in terms of attractiveness to pollinating insects, regardless of other values. The selection of effective pollinators in commercial kiwiberry cultivation can increase yields and produce better-quality fruit. Therefore, further and more detailed studies on the content of volatile compounds in flowers of various species of actinidia and their influence on the behaviour of pollinating insects are needed.

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