Canopy-scale flux measurements and bottom-up emission estimates of volatile organic compounds from a mixed oak and hornbeam forest in northern Italy

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Abstract. This paper reports the fluxes and mixing ratios of biogenically emitted volatile organic compounds (BVOCs) 4 m above a mixed oak and hornbeam forest in northern Italy. Fluxes of methanol, acetaldehyde, isoprene, methyl vinyl ketone + methacrolein, methyl ethyl ketone and monoterpenes were obtained using both a proton-transfer-reaction mass spectrometer (PTR-MS) and a proton-transfer-reaction time-of-flight mass spectrometer (PTR-ToF-MS) together with the methods of virtual disjunct eddy covariance (using PTR-MS) and eddy covariance (using PTR-ToF-MS). Isoprene was the dominant emitted compound with a mean daytime flux of 1.9 mg m\textsuperscript{-2} h\textsuperscript{-1}. Mixing ratios, recorded 4 m above the canopy, were dominated by methanol with a mean value of 6.2 ppbv over the 28-day measurement period. Comparison of isoprene fluxes calculated using the PTR-MS and PTR-ToF-MS showed very good agreement while comparison of the monoterpene fluxes suggested a slight over estimation of the flux by the PTR-MS. A basal isoprene emission rate for the forest of 1.7 mg m\textsuperscript{-2} h\textsuperscript{-1} was calculated using the Model of Emissions of Gases and Aerosols from Nature (MEGAN) isoprene emission algorithms (Guenter et al., 2006). A detailed tree-species distribution map for the site enabled the leaf-level emission of isoprene and monoterpenes recorded using gas-chromatography mass spectrometry (GC–MS) to be scaled up to produce a bottom-up canopy-scale flux. This was compared with the top-down canopy-scale flux obtained by measurements. For monoterpenes, the two estimates were closely correlated and this correlation improved when the plant-species composition in the individual flux footprint was taken into account. However, the bottom-up approach significantly underestimated the isoprene flux, compared with the top-down measurements, suggesting that the leaf-level measurements were not representative of actual emission rates.
1 Introduction

The term volatile organic compound (VOC) describes a broad range of chemical species emitted from natural and anthropogenic sources into the atmosphere. VOCs emitted from the biosphere are commonly termed biogenic VOCs (BVOCs). Of the BVOCs, isoprene is almost certainly the dominant species globally with an estimated annual emission of 535–578 × 10^{12} g C (Arnth et al., 2008; Guenther et al., 2012). Isoprene, along with larger terpenoids, is the BVOC that has received the most attention in the literature to date. Although isoprene is the most commonly measured BVOC, global emission estimates of isoprene continue to differ and there are still large uncertainties associated with the emission estimates of many other compounds. For example, annual monoterpene emission estimates vary between 32 × 10^{12} and 127 × 10^{12} g C (Arnth et al., 2008). A better understanding of how emissions change with land cover, temperature, soil moisture and solar radiation is required to constrain model descriptions of the effects of BVOCs on atmospheric chemistry in the past, present and future (Monks et al., 2009).

BVOCs are a major source of reactive carbon into the atmosphere and as such exert an influence on both climate and local air quality. BVOCs are oxidized primarily by the hydroxyl radical (OH), itself formed by the photolysis of ozone to form peroxide radicals (RO_2). In the presence of NO_x (NO and NO_2) these RO_2 radicals can oxidize NO to NO_2, which may undergo photodissociation leading to the net formation of tropospheric ozone (Fehsenfeld et al., 1992). Tropospheric ozone can then impact human health, forest productivity and crop yields (Royal Society, 2008; Ashmore, 2005). In addition, BVOC species contribute significantly to the formation of secondary organic aerosol (SOA) in the atmosphere. This affects climate both directly and indirectly by the scattering of solar radiation and by acting as cloud condensation nuclei. The formation of cloud condensation nuclei leads to increased cloud cover and therefore an altering of the Earth’s albedo (Hallquist et al., 2009).

The Bosco Fontana campaign was carried out as a part of the ÉCLAIRE (Effects of Climate Change on Air Pollution and Response Strategies for European Ecosystems) EC FP7 project to study the surface–atmosphere exchange within a semi-natural forest, situated within one of the most polluted regions in Europe, and its interaction with air chemistry. During the Bosco Fontana campaign, VOC fluxes and mixing ratios were measured 4 m above the canopy of a semi-natural forest situated in the Po Valley, northern Italy (45°11′51″N, 10°44′31″E), during June and July 2012. The Po Valley experiences high levels of anthropogenic pollution caused by its proximity to the city of Milan’s high levels of industrial and traffic-related pollutant emissions, intensive agriculture and periods of stagnant air flow caused by the Alps to the north and west and the Apennines to the south (Bigi et al., 2011; Decesari et al., 2014).

In order to make accurate air quality predictions, precise regional and global models of BVOC emission are necessary. The modelling of BVOC emission at regional and global scales is generally dependent upon species-specific emission factors for the BVOCs of interest (Guenther et al., 2006; Steinbrecher et al., 2009). These emission factors are usually determined by the measurement of BVOC emission at a leaf level and at standard conditions (generally a leaf temperature of 30°C and 1000 μmol m^{-2} s^{-1} PAR). It is, however, important that leaf-level BVOC emission factors accurately represent canopy-scale emissions. Here we report the fluxes and mixing ratios of a range of BVOCs recorded from mixed mesophore forest at the Bosco Fontana field site. We compare BVOC flux calculation from above-canopy eddy covariance measurements using both a proton-transfer-reaction mass spectrometer (PTR-MS) and a proton-transfer-reaction time-of-flight mass spectrometer (PTR-ToF-MS) with isoprene and monoterpenes fluxes obtained by scaling up leaf-level emission factors using the Model of Emissions of Gases and Aerosols from Nature (MEGAN) to produce a canopy-scale bottom-up modelled flux estimate. We further explore the potential of accounting for the spatial tree-species distribution for improving the comparison between top-down and bottom-up approaches, in what we believe is the first approach of its type.

2 Methods

2.1 Site description

Measurements were taken at a site within the Bosco Fontana Nature Reserve (45°11′51″N, 10°44′31″E), a 233 ha area of semi-natural woodland situated in the municipality of Marmirolo in the Po Valley. The forest canopy had an average height of approximately 28 m and was principally comprised of Carpinus betulus (hornbeam) and three oak species Quercus robur (pedunculate oak), Quercus cerris (turkey oak) and the introduced Quercus rubra (northern red oak) (Dalponte et al., 2007). In the centre of the forest there was a cleared area containing a seventeenth-century hunting lodge surrounded by hay meadows. The surrounding area was predominantly arable farmland with some pastures to the north and west and a reservoir to the north-west. The city of Mantova lies approximately 5 km to the south-east, with the small towns of Marmirolo, Soave and San’Antonio approximately 2 km north, 1 km west and 3 km east respectively. A 42 m measurement tower was situated near the centre of the forest to the south-west of the central hay meadows. The measurement tower was ca. 760 m from the edge of the forest in the direction of the easterly wind that dominated during this measurement period.
2.2 PTR-MS and PTR-ToF-MS set-up and measurement procedure

In order to record BVOC fluxes and concentrations, both a high-sensitivity PTR-MS (Ionicon Analytik GmbH, Innsbruck, described in detail by Blake et al., 2009; de Gouw and Warneke 2007; Hansel et al., 1995; Lindner et al., 1998) and a high-resolution PTR-ToF-MS (Ionicon Analytik GmbH, Innsbruck, as described by Graus et al., 2010; Jordan et al., 2009) were used, together with a sonic anemometer (Gill HS, Gill Instruments Ltd, UK). The PTR-MS was equipped with a quadrupole mass analyser and three turbo molecular pumps (Varian). The Silcosteel inlet and internal tubing were heated to avoid condensation of BVOCs onto internal surfaces. The application of PTR-MS to atmospheric measurements has previously been described by Hewitt et al. (2003) and Hayward et al. (2002).

The sonic anemometer was situated 32 m above the ground on the north-west corner of the tower. This measurement height was chosen due to the fetch restrictions. Fluxes of sensible heat and momentum were compared with a flux measurement at the top of the tower (42 m) and were on average 15 % larger for sensible heat and 5 % for momentum (Finco et al., 2016). It is unclear whether this reflects differences in fetch, instrumentation or the effect of measuring within the surface roughness layer, but it is possible that fluxes reported here are slightly overestimated for this reason. The angle of attack was uncorrelated with wind direction suggesting that there was no local influence on the wind flow. Both the PTR-MS and the PTR-ToF-MS were housed in an air-conditioned cabin at the base of the tower. The PTR-MS subsampled via a ca. 10 cm, 1/8 inch (O.D.) PTFE tube (I.D.: 1 mm, flow rate: 300 mL min⁻¹, residence time: 0.04 s) and with a Reynolds number inside the tube of ca. 258, indicating laminar flow but the very short residence time means that this does not provide the limiting factor for the overall response time of the measurement system) from a 0.5 inch O.D. PTFE common inlet line (I.D. 3/8 inch), heated to avoid condensation, which led from ca. 10 cm below the sonic anemometer to the cabin. Solenoid valves were used to switch between the sample line and zero air, which was generated by passing ambient air through a glass tube packed with platinum catalyst powder heated to 200 °C. The PTR-ToF-MS subsampled via a 3-way valve from the common inlet line; 0.5 L min⁻¹ was pumped through a 1/8 inch (O.D.) and 1/16 inch (O.D.) capillary (together ca. 20 cm long), with 30 mL min⁻¹ entering the instrument and the remaining flow being sent to an exhaust. The common inlet line had a flow rate of ca. 63 L min⁻¹, giving a Reynolds number of ca. 9700 which indicates a turbulent flow. There was no observable influence of the high flow rate on readings from the sonic anemometer, even during periods of relatively low turbulence. Data from both the PTR-MS and the sonic anemometer were logged onto a laptop using a programme written in LabVIEW (National Instruments, Austin, Texas, USA).

The PTR-MS was operated continuously throughout the measurement campaign with interruptions for the tuning of the instrument and refilling of the water reservoir. PTR-MS settings were controlled so that the reduced electric field strength (E/N, where E is the electric field strength and N the buffer gas density) was held at 122 Td (1.22 × 10⁻¹⁹ V m⁻²), with drift tube pressure, temperature and voltage maintained at 2.1 mbar, 45 °C and 550 V respectively. The primary ions and the first water cluster were quantified indirectly from the isotope peaks at m/z 21 (H₃O⁺) and m/z 39 (H₂¹⁸O.H₂O⁺) respectively. The inferred count rate of H₂O⁺ ions over the course of the campaign varied between 1.33 × 10⁶ and 9.00 × 10⁶ counts s⁻¹. O₂⁺ (m/z 32) was kept below 1 % of the primary ion count throughout the campaign in order to limit ionization of VOCs through charge transfer reactions with O₂⁺ and minimize the contribution of the O₂⁺ isotope (¹⁶O¹⁷O⁺) to m/z 33.

During PTR-ToF-MS operation the drift tube temperature was held at 60 °C with 600 V applied across it. The drift tube pressure was 2.3 mbar resulting in an E/N of 130 Td.

A more detailed description of the PTR-ToF-MS operation is provided by Schallhart et al. (2016).

The PTR-MS was operated in three modes: the instrument measured zero air for 5 min followed by 25 min in flux mode, 5 min in scan mode and then a final 25 min in flux mode. While in flux mode, 11 protonated masses were monitored sequentially: m/z 21 the hydronium ion isotope, m/z 39 a water cluster isotope and 9 masses relating to VOCs: m/z 33, 45, 59, 61, 69, 71, 73, 81 and 137. The mass spectral peaks at m/z 21 and 39 were analysed with a 0.2 s dwell time (τ). For the nine VOC species τ = 0.5 s was used in order to increase the instrumental sensitivity to these masses. This gave a total scan time of 4.9 s and the acquisition of ca. 306 data points in each 25 min averaging period. The response time for this instrument, assessed during previous studies and laboratory tests, is ca. 0.5 s, and dwell times were chosen to match this time in order to minimize overall duty cycle loss due to m/z switching. The uncertainty caused by disjoint sampling was calculated and found to cause a 0.17 % error in the flux estimation (see Supplement information for details).

Identification of the compounds observed at each of these masses is complicated by the fact that PTR-MS only allows the identification of nominal masses, therefore it is impossible to distinguish between isobaric compounds. As such, there may be more than one compound contributing to each of the measured masses; Table 1 displays the masses monitored and the compounds likely to be contributing to each mass together with the exact masses observed at each unit mass using the PTR-ToF-MS, which has much greater mass resolution than the quadrupole PTR-MS instrument. It was assumed that the dominant contributions at m/z 33, 45, 59, 61, 69, 71, 73, 81 and 137 were from protonated methanol, acetaldehyde (ethanal), acetone (propanone), acetic acid (ethanoic acid), isoprene (2-methyl-
m/z 137) were present in the calibration gas standard, al-
hyde (trichlorobenzene, C₆H₃Cl₃), isoprene (C₅H₈+), acetic acid (C₂H₃O₂+), methacrolein (MACR, 2-methylprop-2-enal), methyl ethyl ketone (MEK, butanone), a monoterpene mass spectral frag-
ketone (MEK, butanone), methyl ethyl ketone (MVK, butenone) and monoterpenes respectively. A further contribution indicated in brackets. At m/z 61 and 71 the sensitivity was calculated from a transmission curve.

| Unit mass (PTR-MS) | Exact mass (PTR-ToF-MS) | Contributing compound(s) | Formula | PTR-MS sensitivity (ncps ppbv⁻¹) |
|---------------------|-------------------------|---------------------------|---------|---------------------------------|
| 21                  | 21.023                  | Water isotope             | H¹⁸O⁺    | –                               |
| 33                  | 32.997                  | Oxygen isotope            | O¹⁷O⁺    | 11.60 (methanol)                 |
| 39                  | 39.033                  | Methanol                  | CH₅O⁺    | –                               |
| 45                  | 44.997                  | Protonated carbon dioxide | C₅H₁₀O₂⁺ | 9.90 (acetaldehyde)             |
| 59                  | 59.049                  | Acetone                   | C₅H₁₂O⁺  | 8.82 (acetone)                  |
| 61                  | 61.028                  | Acetic acid               | C₅H₈O₂   | 8.40 (transmission curve)       |
| 69                  | 69.0699                 | Isoprene                  | C₅H₁₀⁺   | 3.80 (isoprene)                 |
| 71                  | 71.049                  | Methyl vinyl ketone       | C₅H₁₀O⁺  | 5.29 (transmission curve)       |
| 71                  | 71.049                  | Methacrolein              | C₅H₁₀O⁺  |                                  |
| 71                  | 71.085                  | Unknown                   | C₅H₁₁    |                                  |
| 73                  | 73.026                  | Unknown                   | C₅H₁₂O₂⁺ | 5.87 (Methyl ethyl ketone)      |
| 73                  | 73.047                  | Unknown                   | Unknown  |                                  |
| 73                  | 73.065                  | Methyl ethyl ketone       | C₅H₁₀O⁺  |                                  |
| 73                  | 73.065                  | Butanal                   | C₅H₁₀O⁺  |                                  |
| 81                  | 80.997                  | Unknown                   | C₅H₁₂O₂⁺ | 1.59 (α-pinene fragment)       |
| 81                  | 81.033                  | Unknown                   | C₅H₁₀O⁺  |                                  |
| 81                  | 81.070                  | Monoterpene fragment      | C₆H₉⁺    |                                  |
| 81                  | 81.070                  | Hexenal fragment          | C₆H₉⁺    |                                  |
| 137                 | 137.056                 | Unknown                   | Unknown  | 0.16 (α-pinene)                 |
| 137                 | 137.133                 | Monoterpene               | C₁₀H₁₇⁺  |                                  |

Table 1. Unit masses measured using the PTR-MS during the ÉCLAIRE campaign at Bosco Fontana and the exact masses observed using the PTR-ToF-MS. Where the PTR-MS sensitivity was calculated directly from a compound in the calibration standard, this compound is indicated in brackets. At m/z 61 and 71 the sensitivity was calculated from a transmission curve.

1,3-butadiene), methyl vinyl ketone (MVK, butenone) and methacrolein (MACR, 2-methylprop-2-enal), methyl ethyl ketone (MEK, butanone), a monoterpene mass spectral fragment and monoterpenes respectively. A further contribution to m/z 71, recently identified, are isoprene hydroxy hydroperoxides (ISOPOOH, Rivera-Rios et al., 2014). However, the concentrations of this intermediate are small if NOx concentrations are high and therefore are likely to be negligible at this site, where NOx concentrations were large (Finco et al., 2016).

2.2.1 PTR-MS calibration

The PTR-MS was calibrated using a gas standard (Ionicon Analytic GmbH, Innsbruck) containing 17 VOCs at a mixing ratio by volume of approximately 1 × 10⁻⁶ (ca. 1 ppmv). The protonated mass of the VOCs ranged from m/z 31 (formaldehyde, CH₂O⁺) to m/z 181 (1,2,4-trichlorobenzene, C₆H₃Cl₃). Methanol (m/z 33), acetaldehyde (m/z 45), acetone (m/z 59), isoprene (m/z 69), MEK (m/z 73) and the monoterpene α-pinene (m/z 81 and m/z 137) were present in the calibration gas standard, allowing sensitivities to be calculated directly. Due to reduced quadrupole transmission for high masses, monoterpenes were quantified using the fragment ion at m/z 81. For compounds not contained in the gas standard (acetic acid (m/z 61) and MVK and MACR (m/z 71)) empirical sensitivities were calculated. A relative transmission curve was created using the instrumental sensitivities calculated from the masses present in the standard and from this curve, sensitivities for the unknown masses were calculated (Davison et al., 2009; Taipale et al., 2008). Error in calibration using the gas standard was assumed to be below 15 %, whereas relative errors in calibrations using the relative transmission approach are < 30 % (Taipale et al., 2008). The change in instrumental sensitivity from before the campaign to the end of the campaign was +1.9, −2.1, −0.2, −2.1, −0.3 and −0.7 ncps ppbv⁻¹ for methanol, acetaldehyde, acetone, isoprene and methyl ethyl ketone respectively.

2.2.2 PTR-ToF-MS calibration

Background measurements of the PTR-ToF-MS were made up to three times a day using zero air generated by a cus-
tom made catalytic converter. Calibrations were made using a calibration gas (Appel Riemer Environmental Inc., USA) which contained 16 compounds, with masses ranging from 33 to 180 amu. For VOCs not included in the calibration standard, the average instrument sensitivities towards the known C$_x$H$_y$, C$_x$H$_y$O$_z$ or C$_x$H$_y$N$_z$ compound families were used.

### 2.3 Calculation of volume mixing ratios

Mixing ratios by volume were calculated from data generated using the PTR-MS using a programme written in LabVIEW (National Instruments, Austin, Texas, USA). Mixing ratios by volume ($\chi$) were calculated from the raw PTR-MS data (in counts per second, cps) using a method based on those of Taipale et al. (2008) and Tani et al. (2004).

$$\chi_{\text{VOC}} = \frac{I(\text{RH}^+)_\text{norm}}{S_{\text{norm}}},$$  

(1)

where $S_{\text{norm}}$ is the normalized sensitivity and $I(\text{RH}^+)_\text{norm}$ represents the background corrected normalized count rate (ncps) for the protonated compound $R$, which was calculated as shown below.

$$I(\text{RH}^+)_\text{norm} =$$

$$I(\text{RH}^+)(\frac{I_n}{I(\text{H}_3\text{O}^+)+I(\text{H}_2\text{O}+\text{H}_2\text{O})})(\frac{P_{\text{norm}}}{P_{\text{drift}}})$$

$$-\frac{1}{n}\sum_{i=1}^{n} I(\text{RH}^+)_\text{zero,i}$$

$$\left(\frac{I_n}{I(\text{H}_3\text{O}^+)_\text{zero,i}+I(\text{H}_2\text{O}+\text{H}_2\text{O})_\text{zero,i}}\right)(\frac{P_{\text{norm}}}{P_{\text{drift},\text{zero,i}}}),$$  

(2)

where $I(\text{RH}^+)$, $I(\text{H}_3\text{O}^+)$ and $I(\text{H}_2\text{O}+\text{H}_2\text{O})$ represent the observed count rate for the protonated compound $R$, $\text{H}_3\text{O}^+$ and the $\text{H}_2\text{O}+\text{H}_2\text{O}$ cluster respectively. Subscript zero refers to zero-air measurements, $n$ is the number of zero-air measurement cycles and $P_{\text{drift}}$ is the drift tube pressure. The drift tube pressure was normalized to 2 mbar ($P_{\text{norm}}$) and the sum of the primary ion and first water cluster was normalized to a count rate of $10^6$ cps ($I_n$). The compound-specific limit of detection (LoD) was calculated using the method described by Karl et al. (2003):

$$\text{LoD} = 2 \times \frac{\sigma_{\text{Background}}}{S_{\text{VOC}}},$$  

(3)

where $S_{\text{VOC}}$ is the instrumental sensitivity to the VOC and $\sigma_{\text{Background}}$ is the mean background normalized count rate.

### 2.4 Flux calculations from PTR-MS

The 25 min PTR-MS flux files were inspected, and incomplete or disrupted files were removed. BVOC fluxes were then calculated using a programme also written in LabVIEW, based upon the virtual disjunct eddy covariance technique (vDEC) developed by Karl et al. (2002), also termed continuous flow disjunct eddy covariance (Rinne et al., 2008). This method has been successfully applied in a number of studies (e.g. Davison et al., 2009; Langford et al., 2009; 2010a, b; Misztal et al., 2011; Rinne et al., 2007). This approach allows direct calculation of fluxes of atmospheric constituents, as with standard eddy covariance, yet in this case sampling of scalar concentrations is not continuous. The flux, $F_x$, for each compound was calculated using a covariance function between the vertical wind velocity, $w$, and the VOC mixing ratios, $\chi$:

$$F_x(\Delta t) = \frac{1}{N} \sum_{i=1}^{N} w(i - \Delta t/\Delta t_w)\chi(i),$$  

(4)

where $\Delta t$ is the lag time between the PTR-MS concentration measurements and the vertical wind velocity measurements from a sonic anemometer, $\Delta t_w$ is the sampling interval between wind measurements (0.1 s), $N$ is the number of PTR-MS measurement cycles in each 25 min averaging period (typically 306 in our study) and primes represent the momentary deviations from the mean concentration or vertical wind speed (e.g. $w = \overline{w} - \overline{w}$).

VARIATIONS IN TEMPERATURE, PRESSURE AND THE PERFORMANCE OF THE SAMPLE LINE PUMP CAN CAUSE SMALL DEVIATIONS IN $\Delta t$. Therefore these values were calculated using a cross-correlation function between $w'$ and $\chi'$. Lag times were calculated individually for each $m/z$ monitored by the PTR-MS by selecting the absolute maximum value of the covariance function within a 30 s time window (MAX method, Taipale et al., 2010). This analysis resulted in a clear isoprene flux but for most masses a high proportion of the data fell below the limit of detection. These data, especially in the case of acetone, showed a significant amount of flux values with the opposite sign, “mirroring” the true flux. These “mirrored” points occur when the measured flux is of comparable magnitude to the total random error of the system (Langford et al., 2015). As the cross-correlation maximum is likely to be an overestimate when the noise-to-signal ratio is greater than one, these points were substituted with fluxes calculated using a fixed lag time.

A histogram of isoprene lag times calculated using the MAX method is displayed in the Supplement, showing a clear maximum at 7.5 s. Therefore 7.5 s was chosen as the isoprene-fixed lag time and fixed lag times for the other masses were calculated from the isoprene-fixed lag time, accounting for the dwell times of the different compounds in the measurement cycle.

#### 2.4.1 Flux quality assessment and potential losses

In order to assess the quality of each 25 min flux file, the resultant fluxes were subjected to three quality checks following a two-dimensional coordinate rotation, which was applied to correct for tilting of the sonic anemometer (see Ta-
files failing the stationarity check or falling below the threshold were excluded from further analysis. Data falling below this threshold predominantly occurred at night when wind velocity reached a minimum. Detection limits for each 25 min flux file were calculated using a method based on that of Wienhold et al. (1994), as applied by Spirig et al. (2005), where the signal of the flux at the true lag is compared to the background noise of the covariance function. The 95th percentile of the covariance function in the lag range 150–180 s was calculated and flux files falling below this value were labelled as having fallen below the LoD. Finally data points underwent a stationarity test as described by Foken and Wichura (1996), which assessed the stability of the flux across the 25 min averaging period. Data points found to be generated from periods of non-stationarity were also labelled. Flux files in which all three tests were passed and where only the LoD test was failed were included in all further analyses. Files which failed the LoD test were included to prevent a positive bias being introduced to the results. Flux files failing the stationarity check or falling below the u* threshold were excluded from further analysis.

The integral turbulence characteristics were assessed using the FLUXNET criteria described by Foken et al. (2004). The turbulence at the Bosco Fontana field site was well developed with 87% of the data in the first three categories, defined by Foken et al. (2004) as suitable for fundamental research. Less than 1% of the data fell into category 9, characterized as data to be excluded under all circumstances.

The flux losses in the virtual disjunct eddy covariance system were assessed. Loss of flux at frequencies higher than the threshold were assessed using the criteria of Langford et al. (2010a), data points were labelled if the mean friction velocity (u*) over the 25 min averaging period was found to be below 0.15 m s⁻¹. Data falling below this threshold predominantly occurred at night when wind velocity reached a minimum. Detection limits for each 25 min flux file were calculated using a method based on that of Wienhold et al. (1994), as applied by Spirig et al. (2005), where the signal of the flux at the true lag is compared to the background noise of the covariance function. The 95th percentile of the covariance function in the lag range 150–180 s was calculated and flux files falling below this value were labelled as having fallen below the LoD. Finally data points underwent a stationarity test as described by Foken and Wichura (1996), which assessed the stability of the flux across the 25 min averaging period. Data points found to be generated from periods of non-stationarity were also labelled. Flux files in which all three tests were passed and where only the LoD test was failed were included in all further analyses. Files which failed the LoD test were included to prevent a positive bias being introduced to the results. Flux files failing the stationarity check or falling below the u* threshold were excluded from further analysis.

The percentage of flux-averaging periods during which >25% of the flux originated from outside the forest area was also assessed by footprint analysis and found to account for 26% of the data set. As the flux footprint moves with atmospheric stability, fluxes from outside the forest predominantly occurred during night-time conditions when emission rates were very small. Therefore it was not deemed necessary to specifically remove these data prior to further bulk analysis of the data set, although it is recognized that the u* filter criterion removed many of these measurements. A more detailed analysis of the effect of the tree-species composition within the footprint on measured and modelled fluxes is presented below.

2.5 Flux calculations from PTR-ToF-MS

BVOC fluxes were calculated from PTR-ToF-MS data using the eddy covariance (EC) method similar to that described above for the PTR-MS. The PTR-ToF-MS flux analysis differed in that the cross correlation between w' and χ' was calculated using the method described by Park et al. (2013). Whilst in the PTR-MS measurement, the target compounds are predetermined through the measurement cycle, in the PTR-ToF-MS the entire high-resolution mass spectrum can be used to search for compounds that carry a flux. PTR-ToF-MS data were analysed using the TOF Analyzer V2.45 as described by Müller et al. (2013) and TofTools (Junninen et al., 2010). An automated flux identification routine was then used to calculate the average of the absolute cross-covariance functions during a midday period. The maximum value was then automatically selected from the averaged spectrum and checked against the manually selected noise level (10σnoise) to determine whether a flux was present.

The fluxes were filtered using the 70% stationary criteria as presented by Foken and Wichura (1996), as was applied.
plied to the PTR-MS data and corrected for loss of high-
frequency flux Horst (1997). For a more detailed description of the flux calculation from the PTR-ToF-MS see Schallhart et al. (2016).

2.6 Leaf-level GC–MS measurements

A portable gas exchange system equipped with a controlled-
environment 6-cm² broadleaf cuvette (LI6400, Li-COR, Lin-
coln, USA) was used to measure net photosynthetic rate (A) and stomatal conductance (gₛ) at basal conditions of PAR (1000 µmol m⁻² s⁻¹), leaf temperature (30 °C) and a CO₂ concentration (400 ppm) from fully expanded leaves. These conditions were comparable to those observed during the campaign where the average daytime temperature was 29 °C. While the cuvette is capable of reproducing ambient light and temperature conditions, unstable environmental conditions below the canopy make it difficult to achieve steady-state fluxes. BVOC emission was therefore recorded at basal conditions to ensure that steady-state fluxes could be obtained and to enable comparison between different individual measurements. When A reached a steady state, the outlet tube from the leaf cuvette was replaced with a Teflon tube, and the airstream exiting from the cuvette was used to sample BVOCs (according to the methodology in Loreto et al., 2001) by adsorbing them onto a Silcosteel cartridge packed with 200 mg of tenax (Supelco, PA, USA). Tenax is a very hydrophobic and adsorbent material with high thermal stability, generally used for trapping BVOC (Dettmer and Engel-wald, 2002). The flow rate through the leaf cuvette was main-
tained at 500 µmol s⁻¹, and a subsample of 200 mL min⁻¹
(130 µmol s⁻¹) was pumped through the cartridge with an external pump (AP Buck pump VSS-1) for a total volume of 6 L of air. Blank samples of air without a leaf in the cu-
vette were collected every day before and after the BVOC samplings. Finally the cartridges were sealed and stored at 4 °C until analysis.

The cartridges were analysed using a Perkin Elmer Clarus 580 gas chromatograph coupled with a Clarus 560 Mass-Detector and a thermal-desorber Turbo Matrix (Perkin Elmer Inc., Waltham, MA, USA). The gas chromatograph was equipped with an Elite-5-MS capillary column (30 m length, 250 µm diameter and 0.25 µm film thicknesses). The carrier gas was helium. The column oven temperature was kept at 40 °C for 5 min, then increased with a 5 °C min⁻¹ ramp to 250 °C and maintained at 250 °C for 5 min. BVOC were identified using the National Institute of Standards and Technology (NIST) library provided with the gas-chromatography mass spectrometry (GC–MS) Turbomass software. GC peak retention time was substantiated by analysis of parent ions and main fragments of the spectra. Com-
mmercially available reference standards (gaseous standards, Rivoira, Milan, Italy and liquid standards, Sigma Aldrich, Milan, Italy) were used to create the calibration curves and to quantify the emissions. To normalize the BVOC results, the quantities of terpenes collected from the empty cuvette (blanks) were subtracted from the plant emission results.

2.7 Mapping tree-species distribution

Tree-species distribution data were obtained from Dalponte et al. (2007) who used a combination of light detection and ranging (lidar) and hyperspectral data to develop a high-
resolution tree-species distribution map of the Bosco Fontana Nature Reserve.

The overall accuracy (kappa coefficient) of this species map is particularly high (0.89), considering the number of classes (23) and the number of training samples (20 % of the data are used in the training set and 80 % in the test set) per class. The lidar channels provide relatively sparse information for discriminating between tree species, increasing the overall accuracy of the tree-species assignment using the hyperspectral data by only 1 % but the lidar data significantly increase the accuracy of understory and underrepre-
sented classes. The kappa coefficient of the main species is also very high (0.88–0.93) showing the effectiveness of this approach for species classification in a very complex forest with 20 different broad-leaf species, some of which, such as Q. cerris, Q. robur and Q. rubra, belong to the same genus. For a more detailed discussion of the mapping results and methodology see Dalponte et al. (2007, 2008).

3 Results and discussion

3.1 Meteorological conditions

The measurement campaign at Bosco Fontana ran from 1 June to 11 July 2012 (41 days) with data recorded using the PTR-MS from the 13 June to the 11 July 2012. The me-
teorological conditions recorded at the measurement site dur-
ing this period are summarized in Fig. 1, times are reported in central European time (UTC +1) as used throughout this paper. The campaign average flux footprint is displayed in Fig. 2. With the exception of two heavy thunderstorms, the first in the first week of June before measurements began and the second overnight on 6 July, there was no precipi-
tation during the measurement period. During the measure-
ment period, ambient temperature varied from a low of 14 °C to a high of 35 °C, with temperatures lowest early in the campaign. Daily photosynthetically active radiation (PAR) peaked within the range 1890–2105 µmol m⁻² s⁻¹ and the relative humidity during the campaign varied between 29 and 90 %. Winds were generally easterly or north-westerly. For most of the campaign wind speeds were below 3.5 m s⁻¹ but peaked at 5.6 m s⁻¹ on 23 June, with the mean wind speed for the campaign period of 1.6 m s⁻¹.

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3.2 BVOC mixing ratios and fluxes

BVOC fluxes were recorded at the Bosco Fontana site using both the PTR-MS and the PTR-ToF-MS. Unless stated, the results displayed here were calculated from measurements made using the PTR-MS. Data analysis was carried out with the aid of the R openair package (Carslaw and Ropkins, 2012; R Core Team, 2012). For a full discussion of all fluxes and concentrations recorded using the PTR-ToF-MS, see Schallhart et al. (2016).

The mixing ratios of the eight BVOC species measured in flux mode using the PTR-MS are displayed in Fig. 3 and are summarized in Table 2 (for further details, see Fig. S6). These mixing ratios were calculated using the high-frequency flux measurements so the presented mixing ratios are averaged over 25 min. The mixing ratio LoDs, calculated as described above (Karl et al., 2003; Langford et al., 2009; Misztal et al., 2011), were in the same range as those calculated on previous campaigns (Langford et al., 2009; Misztal et al., 2011) and, with the exception of isoprene for which the mixing ratio dropped towards zero at night, the recorded mixing ratios generally remained above their respective LoD.

Table 3 summarizes the flux data recorded during the Bosco Fontana measurement campaign. Wind speeds decreased at night, leading to a large proportion of the night-time data falling below the $u^*$ threshold of 0.15 m s$^{-1}$. Consequently, average emission fluxes of all eight compounds are reported for the daytime period 10:00–15:00 LT as well as for the whole campaign. Large fluxes of $m/z$ 69 and $m/z$ 81 (assigned to isoprene and monoterpenes respectively) were observed and are shown in Fig. 4. Fluxes of $m/z$ 33, 45, 59, 61, 71 and 73 (assigned to methanol, acetaldehyde, acetone, acetic acid, MVK + MACR and MEK respectively) were also observed, but these fluxes were weaker, leading to a high percentage of fluxes failing the LoD check. However, as is described by Langford et al. (2015), when these flux data are averaged to show the average diurnal cycle, it is appropriate to use a combined LoD value appropriate for the same period rather than the LoD attached specifically to each 25 min flux file. It is, however, essential that each individual flux period be processed carefully to avoid the introduction of a bias due to the use of the MAX method of time-lag identification. The LoD for the mean (LoD) decreases with the square root of the number of samples averaged ($N$).
Risoprene fluxes calculated using both instruments show are displayed in Fig. 4 and summarized in Table 3. The ing both the PTR-MS and the PTR-ToF-MS instruments Mediterranean ecosystems, see the Supplement. Each BVOC and comparison made with other temperate and a more detailed discussion of the fluxes and mixing ratios of results. Data flagged for non-stationarity were excluded. For any way and the application has little influence on the re- pounds showing emission, night-time fluxes are close to zero for depositing compounds. For compounds showing emission, night-time fluxes are close to zero anyway and the application has little influence on the results. Data flagged for non-stationarity were excluded. For a more detailed discussion of the fluxes and mixing ratios of each BVOC and comparison made with other temperate and Mediterranean ecosystems, see the Supplement. The fluxes of isoprene and monoterpenes calculated using both the PTR-MS and the PTR-ToF-MS instruments are displayed in Fig. 4 and summarized in Table 3. The isoprene fluxes calculated using both instruments show very good correlation \( R^2 = 0.91 \), slope 1.3 and intercept 0.17 mg m\(^{-2}\) h\(^{-1}\). The monoterpane fluxes, calculated using \( m/z \) 81 with the PTR-MS and \( m/z \) 81.070 with the PTR-ToF-MS show an \( R^2 = 0.50 \). Three additional mass spectral peaks are observed at \( m/z \) 81 in the PTR-ToF-MS: \( 80.92, 80.99 \) and 81.03, however statistically significant fluxes from these peaks could not be calculated using the PTR-ToF-MS. Owing to the lower sensitivity of the PTR-MS at \( m/z \) 81 and the lower sampling frequency of the disjunct sampling protocol (Rinne and Ammann, 2012), the monoterpane flux calculated using this instrument is significantly noisier than the flux calculated using the PTR-ToF-MS. PTR-MS and PTR-ToF-MS mass scans were averaged over a 10-day period (14–24 June). A comparison of these mass scans over the range \( m/z \) 33 to 100 at unit mass resolution is displayed in Fig. 6, with masses reported relative to \( m/z \) 59 (acetone). A good agreement between the PTR-MS and PTR-ToF-MS is seen for all masses, except for \( m/z \) 33 for which the PTR-MS gives a significantly higher signal. As both instruments have comparable sensitivities at this mass (11.6 and ca. 10–12 ncp s ppbv\(^{-1}\) for the PTR-MS and PTR-.

\[
\text{LoD} = \frac{1}{N} \sqrt{\sum_{i=1}^{N} \text{LoD}^2}
\]

(5)

Therefore, while the flux time series of methanol, acetaldehyde, acetone, acetic acid, MVK + MACR and MEK are not presented here, the campaign average diurnal fluxes are shown (Fig. 5). As discussed above, 25 min averaged flux files flagged as below the LoD were included in these diurnal averages. Flux files falling below the 0.15 m s\(^{-1}\) wind speed threshold were also included to prevent the night-time flux being biased high for depositing compounds. For compounds showing emission, night-time fluxes are close to zero anyway and the application has little influence on the results. Data flagged for non-stationarity were excluded. For a more detailed discussion of the fluxes and mixing ratios of each BVOC and comparison made with other temperate and Mediterranean ecosystems, see the Supplement.

Table 2. Summary of the BVOC mixing ratios (ppbv) recorded at 4 m above the forest canopy during the Bosco Fontana measurement campaign and limits of detection (LoD, ppbv), based on 25 min averages.

| \( m/z \) | 33 | 45 | 59 | 61 | 69 | 71 | 73 | 81 |
|-------|----|----|----|----|----|----|----|----|
| Compound | Methanol | Acetaldehyde | Acetone | Acetic acid | Isoprene | MVK + MACR | MEK | Monoterpenes |
| Max | 14.6 | 3.44 | 7.31 | 14.9 | 4.79 | 1.95 | 1.05 | 0.419 |
| Min | 2.13 | < LoD | 1.18 | 0.396 | < LoD | 0.083 | 0.097 | < LoD |
| Mean | 6.16 | 1.46 | 3.24 | 1.92 | 1.07 | 0.506 | 0.454 | 0.198 |
| Standard deviation | 2.52 | 0.67 | 0.91 | 1.09 | 0.80 | 0.28 | 0.21 | 0.07 |
| Median | 5.69 | 1.30 | 3.14 | 1.73 | 0.934 | 0.506 | 0.428 | 0.199 |
| 1st Quartile | 4.19 | 0.964 | 2.68 | 1.22 | 0.409 | 0.325 | 0.311 | 0.140 |
| 3rd Quartile | 7.53 | 1.87 | 3.82 | 2.31 | 1.53 | 1.95 | 0.568 | 0.245 |
| LOD | 0.436 | 0.712 | 0.239 | 0.141 | 0.167 | 0.081 | 0.048 | 0.067 |

Table 3. Summary of the BVOC fluxes (mg m\(^{-2}\) h\(^{-1}\)) recorded during the Bosco Fontana field campaign based on 25 min values. Values in brackets cover the campaign period where data are available from both instruments to enable direct comparison (15 June–6 July and 15–25 June 2012 for isoprene and monoterpenes respectively).

| \( m/z \) | 33 | 45 | 59 | 61 | 69 | 71 | 73 | 81 |
|-------|----|----|----|----|----|----|----|----|
| Compound | Methanol | Acetaldehyde | Acetone | Acetic acid | Isoprene | MVK | MEK | Monoterpenes |
| Max emission flux | 0.492 | 0.436 | 0.585 | 0.328 | 9.867 (9.907) | 9.195 (9.867) | 0.641 | 0.181 |
| Max deposition flux | −1.589 | −0.335 | −0.692 | −0.876 | −0.238 (−0.238) | −0.305 (−0.305) | −0.457 | −0.128 |
| 1st Quartile | −0.032 | −0.011 | −0.029 | −0.044 | 0.005 (0.005) | 0.019 (0.019) | −0.012 | −0.012 |
| 3rd Quartile | 0.070 | 0.053 | 0.057 | 0.033 | 1.624 (1.796) | 2.661 (2.661) | 0.054 | 0.024 |
| Median | 0.017 | 0.024 | 0.016 | −0.007 | 0.961 (1.003) | 1.465 (1.465) | 0.025 | 0.009 |
| Mean | 0.123 | 0.067 | 0.098 | 0.091 | 1.389 (1.387) | 1.911 (1.911) | 0.076 | 0.039 |
| 0.010 | 0.013 | 0.008 | 0.000 | 0.108 (0.199) | 0.410 (0.410) | 0.011 | 0.005 |
| Mean daytime flux (06:00–18:00) | 0.033 | 0.045 | 0.030 | 0.001 | 1.912 (1.978) | 2.917 (2.917) | 0.049 | 0.018 |
| Standard deviation | 0.161 | 0.082 | 0.125 | 0.096 | 1.401 (1.383) | 1.842 (1.842) | 0.095 | 0.050 |
| Median daytime flux (06:00–18:00) | 0.038 | 0.044 | 0.026 | 0.001 | 1.635 (1.790) | 2.905 (2.905) | 0.041 | 0.014 |

Therefore, while the flux time series of methanol, acetaldehyde, acetone, acetic acid, MVK + MACR, and MEK are not presented here, the campaign average diurnal fluxes are shown (Fig. 5). As discussed above, 25 min averaged flux files flagged as below the LoD were included in these diurnal averages. Flux files falling below the 0.15 m s\(^{-1}\) wind speed threshold were also included to prevent the night-time flux being biased high for depositing compounds. For compounds showing emission, night-time fluxes are close to zero anyway and the application has little influence on the results. Data flagged for non-stationarity were excluded. For a more detailed discussion of the fluxes and mixing ratios of each BVOC and comparison made with other temperate and Mediterranean ecosystems, see the Supplement.
terfere with the methanol signal at m/z 33 but as a significant peak is not observed at m/z 34 (O\(^{18}\)O\(^{+}\)) a large contribution from O\(^{17}\)O\(^{+}\) to m/z 33 is unlikely. This suggests that there is a greater formation of O\(^2\)H\(^{+}\) in the PTR-MS than in the PTR-ToF-MS under these particular operation parameters. No major mass spectral peaks are observed in one instrument alone, indicating that there is no artefact formation or unexpected loss of chemical species with either instrument. The mass scans show a much cleaner spectrum than was reported by Misztal et al. (2011) above an oil palm plantation in south-east Asia, suggesting an atmosphere dominated by fewer chemical species at higher concentrations.

### 3.2.1 BVOC correlations

Scatter plots were used to investigate the relationship between the measured species. Methanol, acetone and MEK (Fig. 7) all showed a shift in the regression of the BVOCs with increasing temperature with two linear groupings observed, one at lower temperature (ca. <20°C) and another at higher temperatures (ca. >20°C). The change in regression could be a result of either different proportions of BVOCs present in high- and low-temperature air masses or by two different sources contributing to the mixing ratios (most likely an atmospheric background and a photochemical source at higher temperatures). It is possible that a second compound could contribute to the nominal mass at higher temperatures but as few compounds have been reported to contribute to m/z 33 or 59, this seems unlikely.

### 3.2.2 Short-chain oxygenated BVOCs

A mean methanol mixing ratio of 6.2 ppbv at 4 m above the canopy was recorded over the duration of the campaign, making it the dominant BVOC observed at Bosco Fontana. Large mixing ratios of methanol compared with other VOC species (caused by its low photochemical reactivity) have been reported in urban landscapes (Langford et al., 2009). This suggests that the large methanol mixing ratios relative to other VOCs observed 4 m above the forest at Bosco Fontana may be due to the surrounding agricultural and urban landscape. Mean acetaldehyde, acetone and acetic acid mixing ratios were 3.4, 3.2 and 1.9 ppbv at 4 m above the canopy respectively. Methanol, acetaldehyde and acetic acid mixing ratios all followed similar diurnal cycles (Fig. 3), with mixing ratios remaining stable through the night before a drop in the morning, probably caused by expansion of the planetary boundary layer after sunrise. Then mixing ratios increased again in the late afternoon as emissions accumulated in a shrinking boundary layer. Acetone mixing ratios remained on average stable throughout the day (Fig. 3). This would suggest a daytime source of acetone offsetting the dilution caused by expansion of the planetary boundary layer. As the flux of acetone, where observed, was very small this source must either be photochemical or situated outside the forest.

The flux of methanol peaked at 0.49 mg m\(^{-2}\) h\(^{-1}\) with a mean daytime flux of 0.03 mg m\(^{-2}\) h\(^{-1}\) (Fig. 5). Methanol deposition was observed during the night and mornings followed by a rapid increase in methanol emission in the late morning and peaking in the early afternoon. Bidirectional exchanges of methanol have been reported previously (for example Fares et al., 2012; Karl et al., 2004) with methanol absorption/desorption thought to occur in thin water films within the canopy (Wohlfahrt et al., 2015). The mean morning (06:30–10:30 LT) methanol deposition velocity (V\(_d\)) at the measurement height (z\(_m\)) was calculated using the relationship (Misztal et al., 2011):

\[
V_d = \frac{m}{A \times \rho \times z_m}
\]
Figure 4. Time series of isoprene (top) and monoterpane (middle) fluxes (mg m$^{-2}$ h$^{-1}$) measured using vDEC. 25 min averaged flux data from the PTR-MS which passed all tests (blue circles), fell below the $u_*$ threshold (triangles) and fell below the LoD (diamonds). Red circles and lines represent PTR-ToF-MS isoprene and monoterpane fluxes with 30 min averaged flux files failing the stationarity test removed. Bottom: scatter plot showing the relationship between isoprene fluxes calculated using PTR-MS and PTR-ToF-MS.

\[ V_d(z_m) = -\frac{F}{\chi(z_m)}, \]  

and was found to be 0.31 cm s$^{-1}$. The night-time deposition velocity was lower, 0.02 cm s$^{-1}$, falling at the bottom end of the 0.02–1.0 cm s$^{-1}$ range reported by Wohlfahrt et al. (2015) from a review of eight different Northern Hemisphere sites.

Acetic acid deposition was also observed in the morning, but any emission flux in the afternoon remained below the limit of detection, even if aggregated into mean diurnal cycles. The mean diurnal acetaldehyde flux is shown in Fig. 5. The flux increased from below the detection limit in late morning to a peak in the early afternoon before dropping again towards zero at night. The flux peaked at 0.44 mg m$^{-2}$ h$^{-1}$ on 29 June and the campaign mean daytime flux was 0.06 mg m$^{-2}$ h$^{-1}$. As can be seen in Fig. 5, the acetone flux remained below the limit of detection for most of the day with a small positive flux observed in the late afternoon.

3.2.3 MVK + MACR and MEK

MVK and MACR are the main products formed following the first stage of isoprene oxidation in the atmosphere (Atkinson and Arey, 2003), accounting for ca. 80% of the carbon.
MACR can also be directly produced within plants as a by-product in the production of cyanogenic glycosides (Fall, 2003), and experimental observation demonstrated that emission of MVK and MACR increase with temperature stress (Jardine et al., 2012). The midday (10:00–15:00 LT) mixing ratios of MVK + MACR at 4 m above the canopy showed a positive correlation with those of isoprene ($R^2 = 0.49$), suggesting that the oxidation of isoprene was responsible for the formation of MVK and MACR.

The production of MVK and MACR from isoprene at the Bosco Fontana site has been modelled by Schallhart et al. (2016), who estimated that 4–27% of the MVK + MACR flux was formed from isoprene oxidation products. MVK and MACR mixing ratios recorded at 4 m above the canopy (Fig. 3) increase in the morning as isoprene concentrations rise, before boundary layer expansion causes them to drop in the middle of the day. The mixing ratios then increase again in the evening as the boundary layer contracts. The flux of MVK + MACR (Fig. 5) peaked in the early afternoon with a mean daytime flux of 0.05 mg m$^{-2}$ h$^{-1}$. This flux is comparable to the 0.03 and 0.08 mg m$^{-2}$ h$^{-1}$ observed by Kalogridis et al. (2014) and Spirig et al. (2005) over European oak and mixed forests.

MEK may be directly emitted by plants (Fall, 2003) or formed photochemically (Luecken et al., 2012). MEK mixing ratios 4 m above the forest canopy remained stable through the night at ca. 0.6 ppbv before dropping in the morning, probably caused by expansion of the planetary boundary layer, to ca. 0.3 ppbv and rising again in the evening (Fig. 3). A plot of the mixing ratios of MEK against those of acetone reveals a bimodal distribution suggesting two distinct sinks or sources (Fig. 7), the first occurring at lower temperatures (ca. 12–20 °C) with a MEK to acetone ratio of ca. 0.17 and the second at higher temperatures (ca. 20–34 °C) with a MEK to acetone ratio of ca. 0.06. A relationship between acetone and MEK has been reported by Riemer et al. (1998), who observed an MEK to acetone ratio of 0.07 at temperatures between 20 and 37 °C. This compares well with the observations at Bosco Fontana. This trend was not observed when data were coloured by PAR, indicating that the bimodal distribution is not driven by the faster rate of reaction of MEK than of acetone with OH. A low MEK emission flux was observed in the afternoon with a mean daytime flux of 0.02 mg m$^{-2}$ h$^{-1}$.

**3.2.4 Isoprene and monoterpenes**

Isoprene mixing ratios 4 m above the canopy began to rise in the mid-morning from a night-time zero, peaking in the late afternoon at ca. 2 ppbv before falling again to zero in the late evening (Fig. 3). Isoprene fluxes were not observed at night, but increased in the morning to a peak in the mid-afternoon before dropping to zero again in the evening (Fig. 5) with a mean daytime flux of 1.9 mg m$^{-2}$ h$^{-1}$.

Isoprene fluxes correlated with leaf temperature (estimated using a method based on that described by Nemitz et al. (2009) and explained in more detail in the Supplement, $R^2 = 0.73$ for an exponential fit), PAR ($R^2 = 0.75$ for an exponential fit) and sensible heat flux (H) ($R^2 = 0.67$). The relationship between isoprene fluxes and mixing ratios, temperature and PAR is displayed in Fig. 8. Table 4 compares isoprene flux measurements with the fluxes recorded during other field campaigns in the Mediterranean region and the isoprene emission factor under basal conditions. As would be expected, the flux of isoprene is shown to be highly dependent on ecosystem type. The fluxes observed during this measurement period, when normalized to standard conditions, were lower than those observed over woodland dominated by isoprene-emitting oak species (Baghi et al., 2012; Kalogridis et al., 2014) due to the lower proportion of isoprene-emitting...
species in the canopy but closer in magnitude to that observed over a mixed pine and oak forest (Fares et al., 2013).

The campaign mean monoterpene mixing ratio 4 m above the canopy was 0.2 ppbv. The diurnal profile (Fig. 3) shows a night-time mixing ratio of ca. 0.18 ppbv, which increases to ca. 0.21 ppbv in the morning, remaining stable through the day and dropping again to ca. 0.18 ppbv at night. The monoterpene flux (Fig. 5) peaked in the early afternoon with a campaign mean midday flux of 0.12 mg m$^{-2}$ h$^{-1}$. Monoterpene mixing ratios were not significantly correlated with leaf-surface temperature or with PAR ($R^2 = 0.11$ and 0.12 respectively). However, the flux displayed a correlation with both leaf-surface temperature and PAR ($R^2 = 0.44$ and 0.39 respectively).

### 3.3 Impacts on air quality

The forest at Bosco Fontana provides a large source of BVOCs in a region of predominantly agricultural and urban land use. The oxidation of BVOCs leads to the formation of low-volatility organic compounds which in turn contribute to SOA (Ehn et al., 2014). The importance of individual BVOC species to SOA formation is, however, variable, with large and cyclic compounds likely to contribute more to SOA formation (Hallquist et al., 2009). Monoterpene are known to contribute significantly to SOA formation. The principal monoterpene species observed during this campaign were $\alpha$-pinene, $\beta$-pinene, sabinene and limonene (Table 5). Following ozonolysis of $\alpha$-pinene and $\beta$-pinene, Lee et al. (2006) observed aerosol yields of 41 and 17 %. Aerosol yields of 41 and 17 % were assigned to limonene and sabinene, due to the placement of C–C double bonds within/external to the cyclic structure. The average aerosol yield from monoterpene ozonolysis during the campaign may then be calculated based on the proportion of each compound emitted. This gives a ca. 39 % yield of aerosol, contributing ca. 0.38 µg C m$^{-3}$ to aerosol (based on the campaign average monoterpene mixing ratio) (0.198 ppbv).

Significant aerosol formation from isoprene has been reported in low NO$_x$ environments (Claeys et al., 2004), however, the high NO$_x$ concentrations at the Bosco Fontana Nature Reserve (Finco et al., 2016) make a significant contribution to SOA from isoprene unlikely.

In the presence of NO$_x$, BVOCs can facilitate the formation of tropospheric ozone. As the potential for photochemical ozone formation is five times greater from isoprene than from VOCs emitted following urban anthropogenic activity...
MEGAN calculates isoprene fluxes based on the product of an emission activity factor ($\gamma$), a canopy loss and production factor ($\rho$) and a canopy emission factor ($\epsilon$). Therefore, plotting isoprene flux against $\gamma \times \rho$ enables the calculation of a canopy-specific isoprene emission factor (Fig. 9), giving a value of 1.68 mg m$^{-2}$ h$^{-1}$ at standard conditions (1000 µmol m$^{-2}$ s$^{-1}$ PAR and 303 K) for the campaign period. For the purpose of this work, $\rho$ was assumed to be 0.96. This is supported by Schallhart et al. (2016), who found that between 3 and 5% of isoprene emissions were lost within the canopy at the Bosco Fontana reserve. The emission activity factor, $\gamma$, was calculated using the algorithms described by Guenther et al. (2006). Radiative transfer through the canopy was modelled using the model applied by Müller et al. (2008). This model was based on that of Goudriaan and van Laar (1994) and ambient temperature was recorded 4 m above the canopy. The standard light and temperature conditions for MEGAN canopy-scale emission factors are $\sim$ 1500 µmol m$^{-2}$ s$^{-1}$ and 303 K (Guenther et al., 2006). In order to enable direct comparison between the GC–MS data and literature emission factors, the factor which sets the emission activity to unity at standard conditions ($C_{CE}$) was increased to 1.42. This gave standard light conditions (temperature: 30°C and PAR: 1000 µmol m$^{-2}$ s$^{-1}$).

**3.4 Calculation of isoprene and monoterpane canopy-level emission factors**

Although other approaches do exist, isoprene fluxes are widely modelled using MEGAN (Guenther et al., 2006). MEGAN calculates isoprene fluxes based on the product of an emission activity factor ($\gamma$), a canopy loss and production factor ($\rho$) and a canopy emission factor ($\epsilon$). Therefore, plotting isoprene flux against $\gamma \times \rho$ enables the calculation of a canopy-specific isoprene emission factor (Fig. 9), giving a value of 1.68 mg m$^{-2}$ h$^{-1}$ at standard conditions (1000 µmol m$^{-2}$ s$^{-1}$ PAR and 303 K) for the campaign period. For the purpose of this work, $\rho$ was assumed to be 0.96. This is supported by Schallhart et al. (2016), who found that between 3 and 5% of isoprene emissions were lost within the canopy at the Bosco Fontana reserve. The emission activity factor, $\gamma$, was calculated using the algorithms described by Guenther et al. (2006). Radiative transfer through the canopy was modelled using the model applied by Müller et al. (2008). This model was based on that of Goudriaan and van Laar (1994) and ambient temperature was recorded 4 m above the canopy. The standard light and temperature conditions for MEGAN canopy-scale emission factors are $\sim$ 1500 µmol m$^{-2}$ s$^{-1}$ and 303 K (Guenther et al., 2006). In order to enable direct comparison between the GC–MS data and literature emission factors, the factor which sets the emission activity to unity at standard conditions ($C_{CE}$) was increased to 1.42. This gave standard light conditions (temperature: 30°C and PAR: 1000 µmol m$^{-2}$ s$^{-1}$).

**Table 4.** Non-exhaustive summary of isoprene fluxes recorded in the Mediterranean region and the isoprene emission factor under basal conditions (temperature: 30°C and PAR: 1000 µmol m$^{-2}$ s$^{-1}$).

| Ecosystem                        | Dominant species                  | Season  | Mean daytime isoprene flux (mg m$^{-2}$ h$^{-1}$) | Isoprene emission factor under basal conditions (mg m$^{-2}$ h$^{-1}$) | Reference    |
|----------------------------------|-----------------------------------|---------|-------------------------------------------------|----------------------------------------------------------------------|--------------|
| Mixed oak and hornbeam forest    | *Carpinus betulus*               | Summer  | 2.6                                              | 1.7                                                                  | This study   |
| Oak forest                       | *Quercus robur*                   | Spring  | 2.8                                              | 7.4                                                                  | Kalogridis et al. (2014) |
| Oak forest                       | *Quercus pubescens*              | Summer  | 5.4–10.1                                        | 5.4                                                                  | Baghi et al. (2012) |
| Mixed oak and pine forest        | *Pinus pinea*                    | Autumn  | ca. 0.13                                         | 0.61                                                                 | Fares et al. (2013) |
|                                  | *Quercus ilex*                   |         |                                                  |                                                                       |              |
|                                  | *Quercus suber*                  |         |                                                  |                                                                       |              |

**Table 5.** Leaf-level isoprene and monoterpane emission (mg m$^{-2}$ h$^{-1}$) recorded using GC–MS from single leaves under basal conditions (temperature: 30°C and PAR: 1000 µmol m$^{-2}$ s$^{-1}$). ND signifies not detected.

| Tree species                  | Isoprene flux (standard error) | $\alpha$-pinene flux (standard error) | Sabinene flux (standard error) | $\beta$-pinene flux (standard error) | Limonene flux (standard error) | Sum monoterpane flux (standard error) |
|------------------------------|-------------------------------|--------------------------------------|-------------------------------|--------------------------------------|-------------------------------|--------------------------------------|
| *Carpinus betulus*           | $2.25 \times 10^{-3}$ (1.50 $\times 10^{-3}$) | $1.07 \times 10^{-2}$ (6.00 $\times 10^{-3}$) | $1.81 \times 10^{-2}$ (1.36 $\times 10^{-2}$) | $5.14 \times 10^{-2}$ (1.23 $\times 10^{-2}$) | $5.83 \times 10^{-1}$ (2.36 $\times 10^{-1}$) | $6.63 \times 10^{-1}$ (2.95 $\times 10^{-1}$) |
| *Quercus robur*              | $2.39 \times 10^{0}$ (6.12 $\times 10^{-1}$) | $2.81 \times 10^{-2}$ (1.45 $\times 10^{-2}$) | ND                            | $4.70 \times 10^{-3}$ (3.08 $\times 10^{-3}$) | $2.16 \times 10^{-1}$ (6.49 $\times 10^{-2}$) | $2.49 \times 10^{-1}$ (9.14 $\times 10^{-2}$) |
| *Quercus rubra*              | $9.14 \times 10^{-1}$ (2.02 $\times 10^{-1}$) | ND                                   | ND                            | $7.95 \times 10^{-3}$ (2.22 $\times 10^{-3}$) | $2.34 \times 10^{-2}$ (7.11 $\times 10^{-3}$) | $3.13 \times 10^{-2}$ (1.11 $\times 10^{-2}$) |
| *Corylus avellana*           | $4.97 \times 10^{-4}$ (3.93 $\times 10^{-4}$) | $1.30 \times 10^{-2}$ (8.00 $\times 10^{-3}$) | ND                            | $2.08 \times 10^{-2}$ (4.80 $\times 10^{-3}$) | $7.57 \times 10^{-1}$ (4.15 $\times 10^{-1}$) | $7.90 \times 10^{-1}$ (2.95 $\times 10^{-1}$) |
| *Acer campestre*             | $4.40 \times 10^{-4}$ (3.11 $\times 10^{-4}$) | $5.14 \times 10^{-2}$ (2.95 $\times 10^{-2}$) | ND                            | $2.27 \times 10^{-1}$ (3.54 $\times 10^{-2}$) | $1.07 \times 10^{-1}$ (1.41 $\times 10^{-2}$) | $3.85 \times 10^{-1}$ (1.11 $\times 10^{-2}$) |
| *Sambucus nigra*             | $4.09 \times 10^{-3}$ (3.66 $\times 10^{-3}$) | ND                                   | ND                            | $9.67 \times 10^{-3}$ (2.69 $\times 10^{-3}$) | $2.49 \times 10^{-1}$ (1.41 $\times 10^{-1}$) | $2.59 \times 10^{-1}$ (2.95 $\times 10^{-1}$) |
| *Cornus sanguinea*           | $4.00 \times 10^{-1}$ (4.00 $\times 10^{-1}$) | $1.11 \times 10^{-3}$ (1.11 $\times 10^{-3}$) | ND                            | $1.95 \times 10^{-2}$ (4.91 $\times 10^{-3}$) | $2.28 \times 10^{-1}$ (1.73 $\times 10^{-1}$) | $2.49 \times 10^{-1}$ (1.11 $\times 10^{-1}$) |
The relationship between temperature (°C) and isoprene fluxes (mg m\(^{-2}\) h\(^{-1}\)) and mixing ratios (ppbv), coloured according to the magnitude of photosynthetically active radiation (µmol m\(^{-1}\) s\(^{-1}\)).

Figure 8. The relationship between temperature (°C) and isoprene fluxes (mg m\(^{-2}\) h\(^{-1}\)) and mixing ratios (ppbv), coloured according to the magnitude of photosynthetically active radiation (µmol m\(^{-1}\) s\(^{-1}\)).

The emission factor is lower than those calculated by Kalogridis et al. (2014) and Baghi et al. (2012) from oak (Quercus pubescens)-dominated forests in southern France (7.4 and 5.4 mg m\(^{-2}\) h\(^{-1}\) respectively). However, this is to be expected, owing to the high proportion of low or non-isoprene-emitting species such as Carpinus betulus, Corylus avellana, Sambucus nigra and Acer campestre present in the forest at Bosco Fontana.

Monoterpene emission from plants may take the form of pool or de novo emission. Emission from stored pools is temperature controlled whereas de novo is driven by photosynthesis and is therefore controlled by light as well as temperature (Ghirardo et al., 2010). Emission from stored pools was modelled using the monoterpene–temperature relationship described by Guenther et al. (1995), this model correlated well with the observed monoterpene flux (PTR-ToF-MS) giving \(R^2\) value of 0.55. In order to assess the effect of light on monoterpene emission, the residual values from the temperature-only model were plotted against PAR (Fig. 10). The residuals displayed a correlation with PAR (\(R^2 = 0.45\)), indicating that light as well as temperature have a significant impact on monoterpene emissions.

Figure 9. Measured isoprene fluxes against the product of \(\gamma\) (emission activity factor, itself the product of the temperature, light and leaf area index activity factors) and \(\rho\) (the canopy loss and production factor).

Figure 10. Plot of the residual values from the temperature-only monoterpene emission model against PAR, demonstrating that light as well as temperature has a significant impact on monoterpene emissions.

from the forest canopy and therefore a significant proportion of monoterpene emission represent de novo emission. However, in order to accurately assess the contribution of pool and de novo emissions to the canopy-scale monoterpene flux, a species-specific leaf-level investigation would be required. A monoterpene canopy emission factor calculated using the MEGAN algorithms, which only simulate de novo emission, was found to be 0.14 mg m\(^{-2}\) h\(^{-1}\).
3.5 Speciated bottom-up isoprene and monoterpene flux estimates derived from leaf-level measurements

Tree-species distribution data combined with information on leaf-level isoprene and monoterpene emission rates and meteorological data were used to produce a bottom-up estimate of the total canopy-level flux. Tree species distribution data were obtained from Dalponte et al. (2007), this tree species distribution map reveals an uneven distribution of isoprene-emitting species within the forest canopy, with the two main isoprene-emitting species (*Q. robur* and *Q. rubra*) concentrated in the south of the forest.

Leaf-level isoprene and monoterpene emissions from the dominant tree species were recorded using GC–MS (Table 5). Together these species represent 76.6% of the total vegetation cover. Isoprene emission was dominated by *Q. robur* and *Q. rubra* with *C. avellana* and *C. betulcus* as the highest monoterpene-emitting species. The isoprene emission recorded for both oak species was lower than that previously reported (Karl et al., 2009; Keenan et al., 2009). For species where GC–MS data were not available, literature values were used. Leaf-level emission factors for minor species for which no GC–MS measurements were made were taken from Karl et al. (2009) with the exception *Rubus* sp. (Owen et al., 2001) and *Acer negundo* and *Morus* (Benjamin et al., 1996). Emission factors taken from the literature were converted from µg g⁻¹ DW h⁻¹ to mg m⁻² h⁻¹ using the mean leaf mass to area ratio, 115 gDW m⁻², reported by Niinemets (1999) from a study of ca. 600 species. The leaf-level emission data were then scaled up to a canopy level using the MEGAN algorithms (Guenther et al., 2006) and incorporated measured PAR and temperature values averaged over 30 min as well as a single-sided leaf-area index (LAI, m² m⁻²) of 5.5.

The hyperspectral/lidar data of Dalponte et al. (2007) were remapped onto a grid centred on the measurement site, with a resolution of 5 m², providing fractional ground cover by each of the 20 tree species within each grid cell. The contribution of each grid cell to each 25 min flux measurement was then calculated at 5 m² resolution using a high-resolution 2-D footprint model based on Kormann and Meixner (2001) similar to that described by Nefte et al. (2008). Finally, the MEGAN algorithm was applied to all plant species using the 25 min meteorology. The information was combined to provide a bottom-up estimate of the flux that the canopy-scale measurements should have detected, based on the leaf-level data. This footprint and species-dependent bottom-up flux estimate showed significantly better agreement with the measured isoprene flux ($R^2 = 0.75$, slope = 0.56) than was observed when the canopy-scale isoprene emission factor calculated above was used ($R^2 = 0.65$, slope = 0.76). This demonstrates the large effect an uneven distribution of isoprene sources can have on the above-canopy flux, even within what appears to be a uniform canopy, as well as the benefit for accounting for spatial species distributions in uniform vegetation canopies.

However, despite capturing the shape of the flux time series, the bottom-up flux underestimated the magnitude of the flux, capturing 56% of the isoprene flux as measured by vDEC. This could in part be caused by changes in vegetation cover between the tree distribution mapping in 2008 and the flux measurements in 2012. There are anecdotal reports that *Populus* sp. coverage has increased in the understory vegetation but it is unlikely that, despite their high rates of growth, the *Populus* coverage changed significantly in the four years between mapping and this campaign. Since 2008, the non-native *Q. rubra* is gradually being removed from the forest. However, this does not explain the discrepancy between the vDEC isoprene flux measurements and the bottom-up flux estimate as the reduction in the number of *Q. rubra* trees should have decreased the flux. Whilst the hyperspectral/lidar tree-species data for this site provide a unique opportunity for comparing the canopy-scale measurements with a detailed bottom-up estimate, the hyperspectral/lidar data only provide information on projected tree species area as seen from above, whilst the flux is regulated by leaf mass and its exposure to radiation. Thus there are uncertainties in the ability of the hyperspectral/lidar to detect understory vegetation and a single conversion factor was used between projected tree area and leaf mass. However, understory vegetation is less exposed to sunlight, reducing its emission. Indeed, the main reason for the underestimate of isoprene flux is probably that the leaf-level isoprene emission rate recorded from the leaves sampled at ground level (albeit taken at the edge of sun exposed clearings) are not representative of those at the canopy top. Substituting the measured *Q. robur* and *Q. rubra* emission factors with those reported by Karl et al. (2009) caused the bottom-up estimate to give 130% of the measured flux and improved the correlation between bottom-up estimates and canopy-scale measurements further.

The speciated monoterpene flux (calculated using GC–MS data and literature values for species where GC–MS data were not available) also showed good agreement with the above-canopy flux ($R^2 = 0.72$) and captured 57% of the flux. The discrepancy between the magnitude of the speciated monoterpene flux and the above-canopy flux could be partially explained by loss of monoterpenes through within canopy oxidation. Schallhart et al. (2016) investigated the flux loss due to chemical degradation using measured concentrations of ozone and NO₃, together with calculated OH and NO₃ concentrations. They found that 5–20% of the monoterpene flux was lost via degradation (in comparison just 3–5% of the isoprene flux was lost). The bottom-up monoterpene flux estimate may also have been affected by the changes to the tree species distribution in the four years between mapping and this campaign, as discussed above, and by deposition of monoterpenes within the forest canopy.

The contribution of different species to the isoprene and monoterpene fluxes over the course of an example day is
Figure 11. The contribution of individual tree species to the speciated isoprene and monoterpene flux on the 2 July 2012. PAR is displayed as a yellow line, wind direction as a black line and the flux recorded using the PTR-MS as bold black bars.

Table 6. Species-specific isoprene and monoterpene emission factors (for a standard temperature of 30 °C and a PAR value of 1000µmol m⁻² s⁻¹). Values derived from optimizing the leaf-level emission factors to give the best fit with the measured above-canopy isoprene and monoterpene fluxes within the constraints displayed.

| Species              | Isoprene emission factor (mg m⁻² h⁻¹) | Isoprene constraint (mg m⁻² h⁻¹) | Monoterpene emission factor (mg m⁻² h⁻¹) | Monoterpene constraint (mg m⁻² h⁻¹) |
|----------------------|---------------------------------------|----------------------------------|----------------------------------------|-----------------------------------|
| Acer campestre       | 0.00                                  | < 1.0                            | 0.15                                   | < 0.50                            |
| Acer negundo         | 0.00                                  | < 1.0                            | 0.33                                   | < 0.64                            |
| Alnus glutinosa      | 0.01                                  | < 1.0                            | 0.22                                   | < 0.50                            |
| Carpinus betulus     | 0.00                                  | < 1.0                            | 0.57                                   | < 0.63                            |
| Corylus avellana     | 0.00                                  | < 1.0                            | 0.23                                   | < 0.50                            |
| Fraxinus angustifolia| 0.00                                  | < 1.0                            | 0.00                                   | < 0.50                            |
| Juglans nigra        | 0.00                                  | < 1.0                            | 0.12                                   | < 0.50                            |
| Juglans regia        | 0.36                                  | < 1.0                            | 0.15                                   | < 0.50                            |
| Morus sp.            | 0.00                                  | < 1.0                            | 0.19                                   | < 0.50                            |
| Platanus hispanica   | 2.97                                  | < 4.4                            | 0.50                                   | < 0.50                            |
| Populus × canescens  | 10.66                                 | < 16.1                           | 0.29                                   | < 0.50                            |
| Populus × hybrida    | 8.06                                  | < 16.1                           | 0.00                                   | < 0.50                            |
| Prunus avium         | 0.00                                  | < 1.0                            | 0.01                                   | < 0.50                            |
| Quercus cerris       | 0.02                                  | < 1.0                            | 0.07                                   | < 0.50                            |
| Quercus robur        | 7.46                                  | <16.1                            | 0.19                                   | < 0.50                            |
| Quercus rubra        | 1.38                                  | < 8.1                            | 0.02                                   | < 0.50                            |
| Robinia pseudoacacia | 1.38                                  | < 2.8                            | 0.01                                   | < 0.50                            |
| Rubus sp.            | 0.00                                  | < 1.0                            | 0.01                                   | < 0.50                            |
| Tilia sp.            | 0.00                                  | < 1.0                            | 0.00                                   | < 0.50                            |
| Ulmus minor          | 0.01                                  | < 1.0                            | 0.01                                   | < 0.50                            |
| Grass                | 0.06                                  | < 1.0                            | 0.06                                   | < 0.15                            |
| Not woodland         | 0.06                                  | < 1.0                            | 0.08                                   | < 0.15                            |
| Outside forest       | 0.06                                  | < 1.0                            | 0.06                                   | < 0.50                            |
shown in Fig. 11. As is shown, the isoprene flux was dominated by *Q. robur* but was sensitive to the species composition within the flux footprint. The change in wind direction around 14:00 LT reduced the contribution of *Q. rubra* to the total flux, with the contribution of *Populus × canescens* increasing significantly. The monoterpane flux was predicted to have been dominated by *C. betuloides*, the dominant tree species in the canopy at Bosco Fontana. A greater number of tree species contributed to the monoterpane flux, and emissions were therefore much more uniform across the canopy and less affected by changes in wind direction.

The fit between the above-canopy measured isoprene and monoterpene fluxes and the bottom-up flux estimate was improved by optimizing the leaf-level emission factors, within the constraints displayed in Table 6, using Chi² minimization as implemented by the solver function in Microsoft Excel. Use of the optimized isoprene and monoterpene emission factors gave good correlations with measured fluxes (R² values of 0.75 and 0.76 respectively). The optimized isoprene and monoterpene emission factors are presented in Table 6 and show a reasonable agreement with literature values (Karl et al., 2009).

4 Conclusions

Direct above-canopy fluxes of methanol, acetaldehyde, acetic acid, isoprene, MVK + MACR, MEK and monoterpenes were calculated using the method of virtual disjunct eddy covariance from mixing ratio data obtained with a PTR-MS above a semi-natural mixed oak and hornbeam forest in northern Italy from 13 June to 11 July 2012. Isoprene was the dominant BVOC emitted with a mean daytime flux of 1.91 mg m⁻² h⁻¹. When normalized to standard conditions (temperature of 30 °C, PAR of 1000 µmol m⁻² s⁻¹) using the MEGAN model (Guenther et al., 2006), a canopy-scale emission factor of 1.68 mg m⁻² h⁻¹ was derived. Mixing ratios of VOCs measured at 4 m above the forest canopy were dominated by those of methanol, with a campaign mean mixing ratio of 6.2 ppbv.

The isoprene fluxes obtained using the PTR-MS/vDEC system showed good agreement with those obtained using a direct eddy covariance (with mixing ratios by volume measured with a fast response PTR-ToF-MS instrument). Monoterpane fluxes recorded using the PTR-MS were noisier and marginally higher than those recorded using the PTR-ToF-MS due to a lower sensitivity and, probably, the inclusion of isobaric compounds. Comparison of mass scan data generated using the PTR-MS and PTR-ToF-MS (m/z 33–100) showed very good agreement with no significant masses observed in one instrument but not in the other.

Upscaling leaf-level isoprene and monoterpene emissions to the canopy scale, using a high spatial resolution tree species database and a 2-D footprint model, showed significantly better correlation with the measured above-canopy fluxes than was obtained using a canopy-scale emission factor. Leaf-level isoprene emission resulted in an underestimate of the above-canopy isoprene flux and this was assumed to be the result of differences in isoprene emission rates from leaves sampled at ground-level and those at the canopy top.

Overall, the data obtained give confidence in the measurement of biogenic VOC fluxes by the method of virtual disjunct eddy covariance and highlight the importance of using leaf-level emission data from sunlit canopy-top leaves when upscaling leaf-level emission to produce a bottom-up canopy-scale emission estimate.

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