Protective Effect of an Exopolysaccharide Produced by Lactiplantibacillus plantarum BGAN8 Against Cadmium-Induced Toxicity in Caco-2 Cells

Emilija Brdarić, Svetlana Soković Bajić, Jelena Đokić, Sladana Đurđić, Patricia Ruas-Madiedo, Magdalena Stevanović, Maja Tolinački, Miroslav Dinić, Jelena Mutić, Nataša Golić and Milica Živković

INTRODUCTION

Cadmium (Cd) is a toxic metal and widespread environmental pollutant with serious adverse effects on human and animal health. In 1993, Cd was classified as a human carcinogen and teratogen (IARC Monographs on the Evaluation of Carcinogenic Risks to Humans, 1993). Cd intoxication has been linked with various diseases, including cancer, diabetes mellitus, cardiovascular diseases, neurodegeneration, and osteomalacia (Genchi et al., 2020). Development of industry, fume inhalation, and use of Cd in paint pigments, Cd–nickel batteries, electroplating, and fertilizers have resulted in high exposure to this metal, with a mortality rate of 17% (International Programme On Chemical Safety, 1992; Nawrot et al., 2010). In addition, its long half-life and low level of excretion make Cd an even more dangerous toxicant. Cd is readily transported from soil to plants with a high bioconcentration factor. In rice, for example, as one of the major staple cereal crops...
for most of the world’s population, the bioconcentration factor of Cd is from 0.300 to 1.112, and it affects the major physiological properties of the plant (Liu et al., 2015). In that context, it is important to mention the so-called “itai-itai” disease that hit Japan during the last century and was a consequence of prolonged intake of Cd-contaminated rice (Huang et al., 2009). An increase of Cd concentration leads to high contamination of the food chain, making food and drinking water the main sources of Cd exposure for the nonsmoking population (Satarug et al., 2010). The first target of orally taken Cd is preferentially the gastrointestinal tract (GIT) (Goon and Klaassen, 1989). Cadmium causes intestinal inflammation, death of epithelial cells, and morphological alterations of cell junctions, which leads to a leaky intestinal barrier (Blais et al., 1999; Prozialeck, 2000; Zhao et al., 2006; Ninkov et al., 2015). After absorption by intestinal epithelium, Cd is transported via blood circulation to different organs and tissues. Thus, keeping the intestinal barrier’s integrity intact is of crucial importance.

In view of all these facts, it is urgent to find novel strategies to prevent and neutralize the toxic effect of Cd. It has been reported that some lactic acid bacteria such as lactobacilli, which are commonly present in the GIT and have GRAS status, can bind toxic metal ions and detoxify them (Halttunen et al., 2008; Mrvcic et al., 2009). These abilities often correlate with the presence of different surface biomolecules [e.g., exopolysaccharides (EPSs)] responsible for the probiotic activity of lactobacilli (Yi et al., 2017). EPSs are carbohydrate polymers, which are covalently or loosely bound to the cell surface or secreted in the cell environment and can have antioxidative and immunomodulatory properties (Kammani et al., 2011; Jones et al., 2014; Caggianiello et al., 2016). Besides, EPS might be used as a postbiotics (Wegh et al., 2019), which are defined as a preparation of inanimate microorganisms and/or their components that confers a health benefit on the host (Salminen et al., 2021). Adsorption of heavy metals by EPSs is mainly a metabolism-independent process based on physicochemical interactions between metal cations and negatively charged acidic functional groups of EPSs (Morillo Pérez et al., 2008). Sequestration of metals is attributable to the presence of various functional groups such as carboxyl, acetate, hydroxyl, amine, phosphate, and sulfate in extracellular bacterial polymers, and the result of metal sequestration may be physical sorption, ion exchange, complexation, and/or precipitation (Gadd and White, 1989; Liu and Fang, 2002). It is now well known that EPS molecules from different bacteria are very potent Cd binders in aqueous solution (Polak-Berecka et al., 2014), but to the best of our knowledge, evidence on how they act in other conditions is insufficient.

### MATERIALS AND METHODS

#### Bacterial Strains, Media, and Growth Conditions

The eight EPS-producing lactobacilli used in this study are listed in Table 1. All strains were grown in De Man–Rogosa–Sharpe medium [MRS (Merck, GmbH, Darmstadt, Germany)]. We prepared MRS agar plates by adding 1.7% agar (Torlak, Belgrade, Serbia). Bacterial strains were grown under anaerobic or aerobic conditions at 30°C or 37°C, depending on the strain.

#### Exopolysaccharide Extraction and Purification

Exopolysaccharide (EPS) molecules were isolated from *Lactiplantibacillus plantarum* BGAN8 (EPS-AN8), the selected strain with the best Cd-binding ability. For EPS isolation, 100 µL of overnight BGAN8 culture was spread on 200 MRS agar plates and incubated for 48 h at 30°C. Isolation of EPSs was done according to the protocol provided by Ruas-Madiedo et al. (2006) with additional steps described by Dinić et al. (2018). Dialysis was performed twice, after the isolation and purification steps, and lasted for 5 days, each time with daily changes of Milli-Q water. The dialysis bag (Sigma–Aldrich, St. Louis, MO, United States) had a 12- to 14-kDa molecular mass cutoff. At the end of dialysis, extracted and purified EPS molecules were lyophilized (Alpha 1-4 LSC plus freeze dryer, Martin Christ, Germany).

#### Analysis of EPS-AN8 Structure by Size Exclusion Chromatography–Multiangle Laser Light Scattering and Testing of Sugar Composition of the Purified Exopolysaccharide

First, the purified EPS was assessed by means of size exclusion chromatography (SEC) coupled with a multiangle laser light

| TABLE 1 | List of strains used in this study. |
|----------|-----------------------------------|
| **Species** | **Strain** | **Origin** |
| *Lactiplantibacillus plantarum* | BGAN8 | Cow white cheese |
| | BGPKM22 | Cow sour milk |
| | BGVL2a-18 | Goat cheese |
| | BGM1 | Cow white cheese |
| | BGSJ2-3 | Cow white cheese |
| *Lactcaseibacillus rhamnosus* | BGH#22 | Human intestinal tract |
| | BGHV954 | Human vaginal tract |
| | BGHV20 | Human vaginal tract |
scattering (MALLS) detector as described by Nikolic et al. (2012). The chromatographic system (Waters, Millford, MA, United States) was composed of an Alliance 2690 module injector, a Photodiode Array PDA 996 detector (set at 280 nm), a 410 refractive index detector, and the Empower software (Waters). The MALLS detector (Dawn Heleos II, Wyatt Europe GmbH, Dambach, Germany) was coupled in series, and the Astra 3.5 software was used for analysis of molar mass distribution. Separation was carried out in two SEC columns placed in series, TSK-Gel G3000 PWXL + TSK-Gel G5000 PWXL, protected with a TSK-Gel guard column (Supelco-Sigma), at a temperature of 40°C and flow rate of 0.45 mL/min using 0.1 M NaNO₃ as the mobile phase. Experiments were repeated three times.

For analysis of neutral sugars in testing EPS monosaccharide composition, polysaccharides (approximately 1.6 mg) were first hydrolyzed with 3M TFA (121°C, 90 min). Monosaccharides were converted into their corresponding alditol acetate by reduction with NaBH₄ and subsequent acetylation (Laine et al., 1972). Identification and quantification were performed by gas–liquid chromatography (GLC) on a gas chromatograph equipped with a DB-5HT column (Agilent, Santa Clara, CA, United States; 30 m × 0.25-mm internal diameter; 0.10-mm film thickness) coupled to a quadrupole mass detector. The oven program started at 175°C for 1 min and was increased by 2.5°C min/min until 204°C was reached. Helium was used as the carrier gas at a flow rate of 1 mL/min. Identification was performed on the basis of coincidence of the retention time of sample components with those previously measured for standards analyzed in identical conditions, using inositol as an internal standard. GLC was performed in the Centro de Investigaciones Biológicas (CIB) by Margarita Salas, CSIC, 28040 Madrid, Spain.

**Fourier Transform Infrared Spectroscopy**

As a powerful analytical technique to investigate the structural characteristics of biomacromolecules (Verhoef et al., 2005), Fourier transform infrared spectroscopy (FTIR) spectroscopy was used to confirm the qualitative composition of EPS-AN8 molecules. The FTIR spectrum of samples was acquired in the transmittance mode on a Nicolet iS10 spectrometer (Thermo Fisher Scientific, Waltham, MA, United States) to confirm the qualitative composition of samples, that is, to confirm that extracted and purified material from BGAN8 is an EPS. Measurements were performed in the spectral range of 400 to 4,000 cm⁻¹ with a resolution of 4 cm⁻¹, the number of scans being 32. Spectra were collected using the attenuated total reflectance mode, whereas the OMNIC software was used to acquire, process, analyze, and manage FTIR data in a graphical environment.

**Preparation of Cadmium Solutions**

Cadmium (Cd) was added in the form of CdCl₂ (Sigma–Aldrich). It was dissolved in Milli-Q water at a concentration of 1 mM and kept at 4°C. Working solutions were freshly made by dissolving CdCl₂ in Milli-Q water or cell culture medium.

**Cadmium-Binding Experiments**

The ability of live EPS-producing lactobacilli to bind Cd was investigated using the protocol provided by Zhai et al. (2013) with slight modifications. Overnight cultures were washed twice in phosphate-buffered saline (PBS). A volume of 1 mL (10⁹ CFU/mL) was taken and centrifuged at 13 000 revolutions/min (rpm) for 10 min at room temperature (RT). Cell pellets were resuspended in 1 mL of dissolved CdCl₂ (50 μM) and incubated at 30°C for 1 h. After 1 h, the mixtures were centrifuged (13 000 rpm, 20 min, RT), and supernatants were collected and kept at −20°C for residual Cd content analysis. The concentration of 50 μM CdCl₂ used in this protocol corresponded to the doses of Cd present in the environment (Blanusa et al., 2002; Damek-Poprawa and Sawicka-Kapusta, 2004).

The EPS of a strain with the highest ability to adsorb Cd ions was isolated, and its adsorption of Cd was measured, following the protocol of Polak-Berecka et al. (2014) with a few modifications. Briefly, 1 mg/mL of EPS was resuspended in Milli-Q water and placed in a dialysis bag. The EPS-containing dialysis bag was placed in a glass cup filled with an aqueous solution of CdCl₂ and stirred for 24 h at 30°C.

**Determination of Cadmium Concentration**

Concentrations of Cd ions were determined by inductively coupled plasma mass spectrometry with ICP-QMS (iCAP Q, Thermo Scientific X series 2, United Kingdom). A standard stock solution of Cd containing of 1,000.0 ± 0.2 mg/L (Alfa Aesar, Germany) was used to prepare intermediate standard solutions for ICP-MS measurements. Operating conditions for ICP-QMS were as follows: RF power—1548 W; lens voltage—7 V; pulse stage voltage—950 V; sample uptake rate—24 rpm.

Table 2. List of primers used for real-time PCR analysis in this study.

| Primer name | Primer sequence | References |
|-------------|----------------|------------|
| IL-8 | 5'-ACACAGAGCTGCAGAATCGAG-3' | Angirianzo et al., 2010 |
| CDH1 | 5'-GGCACAACTTTCAAGAGCAG-3' | Popovic et al., 2020 |
| OCLN | 5'-CACTGAGCACGAAATGTCAC-3' | Elamin et al., 2012 |
| ZO-1 | 5'-AGGGCGCTGTCGTTTCTCTTCAC-3' | Elamin et al., 2012 |
| NQO1 | 5'-AGATGGAGCAGGCTGCGA-3' | This work |
| GAPDH | 5'-GTGAAGGTCCAGATGAGGTC-3' | Freudenberger et al., 2014 |
maintained in the same manner as described by Popović et al. The Caco-2 cells were grown and used as a model system to analyze the adverse effects of Cd and putative protection by the EPS.

**Differentiated human enterocyte-like Caco-2 cells were used**

**Cell Culture**

Differentiated human enterocyte-like Caco-2 cells were used as a model system to analyze the adverse effects of Cd and putative protection by the EPS. The Caco-2 cells were grown and maintained in the same manner as described by Popović et al. (2019). The medium was replaced every second day for 21 days.

**Treatment of Caco-2 Cells**

For all assays, cells were differentiated in 24-well plates, except in the case of the Lucifer yellow test, where cells were differentiated in 24-well plates covered with Transwell inserts (pore diameter 0.4 µm, Sarstedt, Nümbrecht, Germany). Here 21-day cells were washed three times with PBS and then treated. First, in order to find subtoxic concentrations of CdCl₂, cells were treated with CdCl₂ in a range of different concentrations (50, 100, and 200 µM) for 24 h. EPS-AN8 was added in two concentrations: lower and higher, and the incubation lasted another 24 h.

**Cytotoxicity Assay**

The level of lactate dehydrogenase (LDH) released in supernatants, which correlates with the number of dead cells (Bajić et al., 2020b), was measured by using an LDH cytotoxicity assay kit (Thermo Fisher Scientific, Waltham, MA, United States). Activity of LDH was measured in cell culture supernatants, following the manufacturer’s instructions. Absorbance was read at 450 nm using a microplate reader (Tecan Austria GmbH, Grödig, Austria).

**Quantitative Real-Time Polymerase Chain Reaction**

Total RNA was isolated following the protocol described by Sokovic Bajic et al. (2019). For reverse transcription, the RevertAid RT kit was used according to the manufacturer’s protocol (Thermo Fisher Scientific). Amplification of synthesized cDNA was done in a 7500 real-time PCR system (Applied Biosystems, Foster City, CA, United States). The SYBR™ Green PCR Master Mix (Applied Biosystems) was used following the instructions for reaction conditions, viz.: activation at 95°C for 10 min, 40 cycles of 15 s at 95°C, 1:1,000; Invitrogen), anti-claudin (CLDN-4, 1:1,000; Thermo Fisher Scientific), and anti-p65 [nuclear factor κB (NF-κB), 1:500; Cell Signaling Technology] were used during the night at +4°C. Primers used are listed in Table 2. Expression of mRNA was normalized against the GAPDH gene utilizing the 2⁻ΔΔCT method. All primers were purchased from Thermo Fisher Scientific.

**Western Blot**

Protein isolation and Western blotting were performed following the protocol provided by Popović et al. (2019). Isolated proteins were separated on 12% sodium dodecyl sulfate–polyacrylamide gel electrophoresis and transferred to a 0.2-mm nitrocellulose membrane (GE Healthcare). Anti-GAPDH (as loading control; 1:1,000; Invitrogen), anticaludin (CLDN-4, 1:1,000; Thermo Fisher Scientific), and anti-p65 [nuclear factor κB (NF-κB), 1:500; Cell Signaling Technology] were used during the night at +4°C. Proteins were detected by a ChemiDoc apparatus and quantified in ImageJ software (National Institutes of Health).

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**FIGURE 1** Ability of live exopolysaccharide (EPS)–producing lactobacillus strains to bind Cd (A) and ability of EPS isolated from the selected Lactiplantibacillus plantarum AN8 strain (EPS-AN8) to adsorb Cd ions after 1, 2, 3, and 24 h of exposure (B). Values that do not share a common letter are significantly (p < 0.05) different.
Permeability Assay
Integrity of the monolayer barrier was determined by measuring the passive passage of Lucifer yellow dye (Invitrogen, Thermo Fisher Scientific). The protocol described by Yamaura et al. (2016), with a few modifications, was followed. Initially, Lucifer yellow was dissolved in dimethyl sulfoxide and then diluted 1,00-fold in PBS solution. The mixture was added to the apical membrane of Caco-2 cells and incubated for 1 h at 37°C. A microplate reader (Tecan) with excitation and emission of 428 and 536 nm, respectively, was used for fluorescence detection.

Superoxide Dismutase Activity
Superoxide dismutase (SOD) activity was measured in supernatants of Caco-2 cells using the SOD Assay Kit (Sigma–Aldrich) after 24 h of treatments. The manufacturer’s instructions were followed, and absorbance was measured at 440 nm, using a microplate reader (Tecan).

Statistical Analysis
GraphPad Prism 8 software was used in performing statistical analysis and preparing graphs. All experiments were repeated at least three times. After checking normal distribution of values with the Kolmogorov–Smirnov test, one-way analysis of variance and Dunnett and Tukey tests were used for multiple comparison. Values that do not share a common letter are significantly different at $p < 0.05$. Data are presented as mean values ± standard deviation from different experiments.

RESULTS AND DISCUSSION

Ability of Exopolysaccharide-Producing Lactobacilli to Adsorb Cd Ions and Characterization of EPS-AN8
It is known that microorganisms can interact with heavy metals via biosorption of metal ions on the cell surface, intracellular uptake, and chemical transformation (Hassan et al., 2009). Also, it was previously reported that the greatest amount of metals is bound to the surface and adsorbed, in contrast to an exceedingly small part of ions that was actively taken up (Mrvcic et al., 2009). Lactobacilli turned out to be a promising solution for the growing problem of Cd pollution (Mrvčić et al., 2012). They can adsorb a great number of different metals ions, in a strain-specific manner (Zhai et al., 2013). For example, lactobacilli showed a protective effect against acute Cd poisoning in mice. More precisely, both live and dead lactobacilli decreased intestinal absorption of Cd ions, leading to a lower level of Cd in tissues and increased concentration in feces, which indicates the involvement of surface biomolecules of lactobacilli in the Cd-binding process (Zhai et al., 2013). Lactobacilli have the ability to produce homopolysaccharides and heteropolysaccharides with enormous structural diversity and different biological properties (Caggianiello et al., 2016). Among other activities, the potential of different EPS-producing bacterial species to adsorb metal ions was described by Yi et al. (2017). Therefore, in the present study, we tested the potential of eight EPS-producing lactobacilli strains from our laboratory collection to adsorb Cd. L. plantarum BGAN8, L. plantarum BGVL2A-18, and L. plantarum BGPKM22 showed the highest ability to bind Cd ions in aqueous solution and did not significantly differ from each other (Figure 1A). L. plantarum BGAN8 has been well characterized as an EPS producer and microvesicle releaser (Bajic et al., 2020a), and it was therefore chosen for further work. Further, isolated EPS-AN8 was tested for binding Cd. As shown in Figure 1B, EPS-AN8 has a great capacity for binding Cd ions. After the first, second, and third hour of EPS and CdCl$_2$ coincubation, the percentage of Cd binding did not differ significantly, ranging between 72 and 78% of the total amount of CdCl$_2$ in solution. Interestingly, a slight decrease in adsorbed Cd ions was observed after 24 h of coincubation, suggesting that a smaller amount of Cd is reversibly bound to EPS molecules. This observation is in agreement with earlier studies that showed the same phenomenon of a short-term reduction of Cd ion binding between 24 and 48 h (Polak-Berecka et al., 2014). Examination of the SEC profile reveals the presence of one major peak with average retention time of
about 32 min (Figure 2A). Furthermore, SEC-MALLS analyses showed that EPS-AN8 has a high molecular weight (Table 3), the value of which is within the expected range of weight among characterized HePSs of other lactobacilli from the L. plantarum group, ranging from $10^4$ to $10^6$ Da (Tallon et al., 2003; Sánchez et al., 2006; Nikolic et al., 2012). Also, radius of gyration was similar to values for EPS of other lactobacilli (Bachtsarzi et al., 2020). Analysis of neutral sugar content of EPS-AN8 revealed that glucose, galactose, rhamnose, and mannose are present in a ratio of 10.33:0.19:0.05, respectively, and in percentages of 12.91%, 4.23%, 2.45%, and 0.60% (Table 3). It is well known that heteropolysaccharides consist of different monosaccharides, such as D-glucose, D-galactose, and L-rhamnose (De Vuyst and Degeest, 1999). The presence of rhamnose was also reported in other HePSs as well, such as in Lactococcus lactis subspecies cremoris Ropy352, Lactobacillus pentosus LPS26, or Lactobacillus paraplanatarum BGCG11 (Cerning et al., 1994; Knousha et al., 2000; Sánchez et al., 2006). The FTIR spectrum of EPS-AN8 is presented on Figure 2B. The spectrum contains typical groups for polysaccharides, that is, carboxyl, hydroxyl, and amide groups (Dinić et al., 2018). A broad-stretching absorption band at 3,296.62 cm$^{-1}$ corresponds to $\tilde{\nu}$OH or $\tilde{\nu}$NH vibrations (Coates, 2006). From the literature, it is well known that hydroxyl groups are ubiquitous in polysaccharide structure, which exhibits an intense broad stretching vibration in the region characteristic of the carbohydrate ring (Dinić et al., 2018). The affinity of polysaccharides for water molecules depends on the presence of these multi-OH groups (Hu et al., 2017). The small absorption band at 2,925.29 cm$^{-1}$ corresponds to the C-H stretching vibrations of methyl or methylene groups, regularly present in hexoses such as glucose or galactose and in deoxyhexoses such as rhamnose or fucose (Ismail and Nampoththi, 2010). The band at 1,637.24 cm$^{-1}$ represents vibration of the C = O stretch of the amide I band or carboxyl group (Shen et al., 2013). This may indicate the presence in the examined EPS of acidic sugars (monosaccharides with a carboxyl group at one end or both ends of their chain), which are important in view of the heavy metal–binding properties of this polymer (Angyal, 1989). Also, carboxyl and hydroxyl groups are important for the coordination responsible for stability of the EPS-metal bond (Moppert et al., 2009). The band at 1,041.50 cm$^{-1}$ corresponds to a C-O stretch vibration or a phosphorus out-of-phase P-O-C stretch (Corbridge, 2007). EPSs are known to be composed of carbohydrates (sugar residues) substituted with proteins, DNA, phospholipids, and noncarbohydrate substituents such as acetate, glycerol, pyruvate, sulfate, carboxylate, succinate, and phosphate (Angelin and Kavitha, 2020). The strongest absorption band at 1,041.50 cm$^{-1}$ indicates that the substance is an EPS. A possible explanation for the effective binding of metal cations is that the phosphate group undergoes deprotonation under physiological conditions, which results in negative charges along the phosphate backbone (Pal and Paul, 2008; Ruan et al., 2008). The resulting negative charges tend to be stabilized and neutralized by the binding of metal cations, such as, in our case, Cd, and result in immobilization of the metal. The absorption band at 892.44 cm$^{-1}$ corresponds to vibrations of the glycoside link C-O-C (Nikonenko et al., 2000). Apart from the aforementioned peaks, there are no other peaks that can be observed in the spectrum of our sample.

### Protective Effect of EPS-AN8 on Cd-Induced Inflammation of Caco-2 Cells

In order to investigate the ability of EPS-AN8 to protect differentiated Caco-2 cells from Cd-induced inflammation, we determined the subtoxic concentration of CdCl$_2$ on Caco-2 cells (Figure 3A), which were used as an in vitro model of the intestinal epithelium. Concentrations of CdCl$_2$ higher than 50 $\mu$M were toxic, which is in accordance with published data (Hyun et al., 2007), so 50 $\mu$M of CdCl$_2$ was used as a subtoxic concentration in our further experiments. Intestinal epithelial cells are very important, not only as a physical barrier, but also as a part of innate mucosal immunity—producing antimicrobial molecules, as well as the cytokines and chemokines required for immune response activation (Stadnyk, 1994; Turner, 2009). It has been shown that Cd induces inflammation via overproduction of interleukin 8 (IL-8) (Hyun et al., 2007). IL-8 is a chemotactic cytokine that attracts and activates leukocytes, leading to acute inflammation, which is very important for the resolution of infection (Baggiolini and Clark-Lewis, 1992). On the other hand, in the case of stimuli that cannot be removed and persist as a stimulator of the epithelial barrier and immune response, overexpression of IL-8 has a destructive effect on the local tissue (Baggiolini and Clark-Lewis, 1992). In the present study, we demonstrate that treatment of Caco-2 cells with Cd for 3 h leads to a statistically significant increase in IL-8 gene expression (Figure 3B), but cotreatment, at the same time, with the higher concentration of EPS-AN8 provided protection from this overexpression, and the level of mRNA for IL-8 did not differ statistically from the level in the control cells (Figure 3B). On the contrary, cotreatment with the lower EPS-AN8 concentration did not affect the up-regulation of IL-8 induced by Cd. NF-kB is defined as one of the most significant regulators of inflammation in different types of cells, including intestinal epithelial cells (Mitchell et al., 2016). In addition, there are binding regions in the promoter of the IL-8 coding gene, confirming the regulatory role of NF-kB in IL-8 production (Wu et al., 1997). Importantly, it has been shown that activation of NF-kB is essential for up-regulation of IL-8 in Caco-2 cells treated with Cd (Hyun et al., 2007). Accordingly,

| Table 3 | Physicochemical characteristics of EPS-AN8. |
|---|---|---|
| **Parameters** | **Molecular weight (g/mol = Da)** | 2.27 ± 0.07 $\times$ $10^5$ |
| | **Radius of gyration (nm)** | 86.65 ± 3.46 |
| **Monosaccharides** | % | Ratio |
| Glucose | 12.91 | 21.5 |
| Galactose | 4.23 | 7.1 |
| Rhamnose | 2.45 | 4.1 |
| Mannose | 0.60 | 1 |
we investigated the effect of Caco-2 cotreatment with Cd and EPS-AN8 on activation of NF-κB by estimating protein levels of phosphorylated p65, which represents a transcriptionally active form of NF-κB. As shown on Figures 3C,D, both concentrations of EPS-AN8 neutralized the stimulatory effect of Cd on NF-κB activation, but only the higher concentration of EPS-AN8
reduced IL-8 mRNA levels. These results suggest that EPS-AN8, when applied simultaneously with Cd, most likely decreases Cd-induced inflammation through the sequestration of Cd ions by EPS-AN8, which inhibits entry of Cd into the cells, and imply that with the lower EPS-AN8 concentration, it is not sufficient to bind all available Cd ions responsible for mild stimulation of IL-8.

**Protective Effect of EPS-AN8 on Cd-Induced Disruption of Intercellular Junctions**

Maintenance of the intestinal barrier is very important in the restriction of Cd spreading in the organism (Tinkov et al., 2018). For that reason, the permeability of a differentiated layer of Caco-2 cells was analyzed after exposure of cells to Cd and to Cd and EPS simultaneously. According to the results presented in Figure 4A, Cd strongly increased permeability of the monolayer, but integrity of the monolayer was preserved in the culture cotreated with EPS-AN8. Oral administration of Cd disrupts adherence and tight junctions (TJs) in epithelial surfaces, resulting in amplified Cd absorption (Rusanov et al., 2015). Hence, the primary therapeutic targets are proteins involved in intercellular junctions. E-cadherin is defined as the most sensitive to Cd exposure, as Ca ions are substituted with ions of Cd at the sites of its binding to the cells (Prozialeck, 2000). In connection with this, we investigated the potential of two concentrations of EPS-AN8, simultaneously added to Cd, as a treatment of differentiated Caco-2 cells to neutralize the harmful effects of Cd. After 3 h of treatment, only Cd exposure significantly increased the expression of E-cadherin (CDH1) mRNA (Figure 4B), whereas coexposure to both concentrations of EPS-AN8 molecules retained the control value of expression. These results can be explained by the aforementioned Ca–Cd substitution and sensitivity of this protein to an increased concentration of divalent ions. Further, claudin (CLDN-4), zonulin-1 (ZO-1), and occludin (OCLN) are crucial proteins in forming TJs (Günzel and Fromm, 2012). In our study, the level of mRNAs for ZO-1 and OCLN did not change in response to either treatment compared to the control (Figures 4C,D). In contrast, after 24-h treatment, Cd down-regulated the level of CLDN-4 (Figure 4C), whereas cotreatment with EPS-AN8

![Figure 5](image-url)
in both concentrations inhibited the downregulation of this protein's expression (Figures 4E,F). In light of these results, it can be concluded that EPS-AN8 prevents Cd-induced destruction of the intestinal barrier by protecting against CLDN-4 degradation.

Protective Effect of EPS-AN8 on Cd-Induced Oxidative Stress

It has been repeatedly shown that the main mechanism of Cd-induced toxicity is based on induction of oxidative stress in exposed cells (Genchi et al., 2020). We therefore investigated the activity of two enzymes important for oxidative/antioxidative responses of cells: SOD and NAD(P)H quinone reductase (NQO1). Superoxide dismutase is one of the enzymes essential for the antioxidative balance of cells, catalyzing dismutation of the superoxide anion into hydrogen peroxide and molecular oxygen (Canada and Calabrese, 1989). We were able to demonstrate that after 24 h of treatment of Caco-2 with Cd, SOD activity was significantly higher than in the control and in the case where EPS-AN8 was applied to the cells at the same time with Cd. Both concentrations of EPS were shown to be similarly efficient in inhibiting the induction of oxidative stress by Cd through activation of SOD in Caco-2 cells (Figure 5A). As SOD protects cells from the activity of reactive oxygen species (ROS), higher activity implies higher production of ROS (He et al., 2017; Wang et al., 2018). Based on this, it can be concluded that cells are in a better oxidative state in the presence of EPS-AN8 simultaneously with Cd. In addition, NQO1 is also a key enzyme of antioxidant cell defense, one that catalyzes reduction of quinones and a variety of other substrates (Pey et al., 2019). As in the case of SOD, Cd stimulated the level of mRNA for NQO1, which is consistent with data from the literature (Rusanov et al., 2015). The addition of EPS-AN8 in both lower and higher concentrations at the same time with it protected from this effect of Cd as well (Figure 5B).

EPS-AN8 as a Potential Agent for Alleviation of Cd-Induced Damage

From the results obtained when Caco-2 cells were exposed to Cd and EPS-AN8 at the same time, we conclude that EPS-AN8, in both the lower and higher tested concentrations, provides strong protection against the damage caused by Cd toxicity. We presume that such a protective effect is a consequence of the ability of EPS-AN8 to sequester Cd ions and disable the entry of ions into the cells. It was reported elsewhere that L. plantarum CCFM8610, besides initially sequestering Cd ions, can also reverse damage induced by Cd in HT29 cells and mice (Zhai et al., 2016). To see if EPS-AN8 has a therapeutic effect as well, we analyzed changes in the expression of CLDN-4 and NF-κB when Caco-2 cells were previously exposed to Cd. To be specific, after 24 h of treatment of Caco-2 cells with Cd, EPS-AN8 was added to the culture as a putative agent for alleviation of Cd-induced damage and incubated for the next 24 h. Although we noticed some positive changes after treatment with the lower concentration of EPS-AN8, the higher concentration of EPS-AN8 was far more effective in this study (Figure 6). The higher concentration of EPS-AN8 significantly increased expression of CLDN-4 compared to Cd treatment (Figures 6A,B) and significantly decreased the level of p65 protein expression in comparison with Cd treatment and treatment with a lower concentration of EPS (Figures 6C,D). These results imply that EPS-AN8, besides sequestration of Cd ions, also triggered intestinal cell responses that lead to alleviation of Cd-induced damage. This assumption is in accordance with results of a previous study that demonstrated that EPS molecules might have a strong anti-inflammatory potential in conditions of high inflammation via modulation of Toll-like receptor.
expression and inhibition of mitogen-activated protein kinase and NF-κB in intestinal epithelial cells (Gao et al., 2017). Also, some EPSs act as a stabilizer of intestinal barrier integrity by activating signal transducers, activating transcription of signaling (STAT-3) pathways, and up-regulating TJ proteins, respectively, in cases of intestinal barrier dysfunction (e.g., inflammatory bowel disease, colitis) (Zhou et al., 2018).

CONCLUSION

Our findings indicate an important role of probiotic-derived EPSs in protection against the hazardous effects of Cd on intestinal epithelial cells. To our knowledge, the present study is the first one providing information about EPSs as a potent postbiotic agent against this environmental pollutant and its possible use as a functional food supplement or dairy food additive in areas highly exposed to Cd. This pioneering work calls for further studies analyzing the use of EPSs as a putative therapeutic strategy.

DATA AVAILABILITY STATEMENT

The original contributions presented in the study are included in the article/Supplementary Material, further inquiries can be directed to the corresponding author.

AUTHOR CONTRIBUTIONS

EB performed all the work, analyzed and interpreted the data, and drafted this manuscript. MŽ, JD, and MT planned, designed, and performed some of the experiments and critically revised the manuscript. SSB performed Western blot analysis, participated in EPS extraction, analyzed the data, and interpreted the results. PR-M performed SEC-MALLS analysis, analyzed data and interpreted the results. MS analyzed FTIR spectrum of EPS-AN8 and interpreted the results. MD participated in Caco-2 cell culture experiments and qPCR. JM and SD performed Cd-binding research. MŽ and NG supervised all the work. All authors contributed to the article and approved the submitted version.

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SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fmicb.2021.759378/full#supplementary-material
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