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Nanomolar small-molecule detection using a genetically encoded $^{129}$Xe NMR contrast agent†

Genetically encoded magnetic resonance imaging (MRI) contrast agents enable non-invasive detection of specific biomarkers in vivo. Here, we employed the hyper-CEST $^{129}$Xe NMR technique to quantify maltose (32 nM to 1 mM) through its modulation of conformational change and xenon exchange in maltose binding protein (MBP). Remarkably, no hyper-CEST signal was observed for MBP in the absence of maltose, making MBP an ultrasensitive ‘smart’ contrast agent. The resonance frequency of $^{129}$Xe bound to MBP was greatly downfield-shifted ($\Delta \delta = 95$ ppm) from the $^{129}$Xe(aq) peak, which facilitated detection in E. coli as well as multiplexing with TEM-1 $\beta$-lactamase. Finally, a Val to Ala mutation at the MBP–Xe binding site yielded 34% more contrast than WT, with $^{129}$Xe resonance frequency shifted 59 ppm upfield from WT. We conclude that engineered MBPs constitute a new class of genetically encoded, analyte-sensitive molecular imaging agents detectable by $^{129}$Xe NMR/MRI.

Introduction

The development of molecular probes for quantifying the in vivo distribution of biologically active species (e.g., metabolites, neurotransmitters) is critical to understanding both normal physiology and disease pathologies. Magnetic resonance imaging (MRI) offers excellent spatiotemporal resolution and penetration depth without the use of ionizing radiation. Functional MRI (fMRI) has spurred the development of "smart" contrast agents that are responsive to specific physiological triggers such as metal ions, metabolites, and enzymes. Genetically encoded (GE) contrast agents can be expressed directly in the tissue of interest and tailored to bind specific analytes through biomolecular engineering techniques such as directed evolution. In pioneering work, Shapiro and coworkers evolved bacterial cytochrome P450-BM3 to bind dopamine near the paramagnetic heme, which was designed to decrease $^1$H MRI signal: 89 $\mu$M BM3 enabled detection of 75 $\mu$M dopamine in live rat brain. An alternative contrast approach is $^1$H chemical exchange saturation transfer ($^1$H-CEST), which uses selective radiation to depolarize solute-bound protons, then measures loss of bulk solvent proton signal as protons exchange from solute to solvent (i.e., magnetization transfer). By this method, Oskolkov and coworkers observed MR contrast from 733 $\mu$M human protamine-1, an arginine-rich nuclear protein that was sensitive to pH, phosphorylation state, nucleotides, and heparin. Notably, the low sensitivity of $^1$H MRI has limited the development of GE $T_1$, $T_2$, and $^1$H-CEST contrast agents for molecular imaging.

Nuclear hyperpolarization provides a unique strategy to overcome natural limitations in NMR sensitivity. For example, the spin-1/2 nucleus of $^{129}$Xe can be hyperpolarized (hp) through spin-exchange optical pumping (SEOP) to increase net magnetization by several orders of magnitude. Synthetic xenon biosensors have been developed by targeting a cryptophane-A or curcurbit[6]uril host to report on proteins, glycans, oligonucleotides, metal ions, free thiols, and pH. A technique known as hyper-CEST can be employed to detect the hp $^{129}$Xe-host interaction at nM–FM concentrations by applying saturation pulses at the $^{129}$Xe-host chemical shift and monitoring magnetization transfer to the large population of $^{129}$Xe nuclear spins in aqueous solution. This saturation transfer process accelerates the loss of hp $^{129}$Xe(aq) signal relative to the natural depolarization rate, 1/$\tau_T$, which provides an ultrasensitive mechanism for generating MR contrast.

Shapiro et al. demonstrated the first GE hyper-CEST agents via pH detection of proteinaceous nano-to-micron-scale gas vesicles. It remains challenging, however, to develop monomeric protein reporters, as Xe typically binds proteins weakly ($K_a \ll 200$ M$^{-1}$) and with fast exchange ($k_{off} > 10^5$ s$^{-1}$). Recently, our laboratory identified TEM-1 $\beta$-lactamase (Bla) as a monomeric protein reporter, with sub-micromolar Bla producing hyper-CEST signal in mammalian cells. Here, we developed maltose binding protein (MBP) as a small-molecule-responsive, GE xenon biosensor capable of detecting nanomolar concentrations of maltose using the hyper-CEST $^{129}$Xe NMR technique (Scheme 1).

Xe binding to MBP has been well-characterized by NMR, which showed that $^{129}$Xe chemical shift depends on MBP conformation. X-ray crystallography identified a single Xe-
binding site near the maltose binding cleft of MBP (Fig. 1 and S1). MBP is a periplasmic protein encoded by the malE gene that serves as an initial receptor in the maltose/maltodextrin transport systems of Gram-negative bacteria. WT MBP binds maltose with good affinity (\(K_d \approx 1 \mu M\)), as well as other maltodextrins, between two nearly symmetrical lobes that transition from an “open” (MBP\text{open}) to “closed” (MBP\text{closed}) conformation upon ligand binding. This structural response to ligand binding has led to the utilization of MBP as a versatile platform for biosensing applications. Small-molecule detection with MBP, as well as other periplasmic binding proteins (PBPs), has been achieved through a variety of signal transduction modalities, including fluorescence, fluorescence resonance energy transfer (FRET), and electrochemical response. Moreover, MBP has been engineered to increase ligand affinity and to bind non-maltodextrin ligands such as sucrose and zinc. Thus, we set out to evaluate the hyper-CEST NMR contrast generated by MBP as a function of maltose binding.

Results and discussion

Maltose detection by MBP

Xenon hyper-CEST z-spectra were acquired from recombinant MBP in both the presence and absence of maltose to assess the magnitude and frequency of NMR saturation contrast (Fig. 2a). Multiple selective d-SNOB saturation pulses were scanned over the chemical shift range of \(-100\) to \(180\) ppm in 5 ppm steps, and the \(^{129}\text{Xe\text{aq}}\) signal was measured as a function of saturation pulse offset. MBP in the absence of maltose (MBP\text{open}) showed a single saturation response corresponding to free \(^{129}\text{Xe}\) in solution centered at 0 ppm. In contrast, MBP in the presence of maltose (MBP\text{closed}) showed a pronounced saturation response 95 ppm downfield of the \(^{129}\text{Xe\text{aq}}\) peak, corresponding to \(^{129}\text{Xe\text{@MBP\text{closed}}}\). This peak is 35 ppm further downfield than \(^{129}\text{Xe\text{@Bla}}\), suggesting that \(^{129}\text{Xe\text{@MBP\text{closed}}}\) experiences a more hydrophobic environment. Also, the width of the \(^{129}\text{Xe\text{@MBP\text{closed}}}\) peak (35 ppm) is narrower than \(^{129}\text{Xe\text{@Bla}}\) (60 ppm), indicating slower Xenon exchange with MBP (Table S1).

To assess the detection sensitivity of MBP, time-dependent saturation transfer experiments were performed by measuring \(^{129}\text{Xe\text{aq}}\) polarization as a function of saturation time (Fig. S2). Saturation frequencies of d-SNOB pulses were positioned +95 ppm and −95 ppm, referenced to the \(^{129}\text{Xe\text{aq}}\) peak, for on- and off-resonance, respectively. The normalized difference

### Scheme 1

Ultra-sensitive detection of a small molecule (maltose)-protein (MBP) interaction via hyper-CEST NMR. HP \(^{129}\text{Xe}\) (green) binds maltose-bound MBP, where the unique \(^{129}\text{Xe}\) resonance frequency is saturated by shaped RF pulses. \(^{129}\text{Xe}\) exchange leads to depolarization of solution-phase \(^{129}\text{Xe}\) pool, thereby generating MR contrast (yellow peak).

### Fig. 1

Xe (red sphere) bound to MBP\text{open} (PDB ID 1LLS), with the N-terminal domain colored blue, C-terminal domain colored green, and linking segments colored orange. (Inset) detailed view of the Xe-binding cavity.

### Fig. 2

(a) Hyper-CEST z-spectra of 80 \(\mu M\) MBP with and without 1 mM maltose in pH 7.2 PBS at 300 K, with z-spectra of buffer only with and without maltose shown for reference. Pulse length, \(\tau_{\text{pulse}} = 3.8029\) ms; field strength, \(B_{1,\text{max}} = 77 \mu T\). (b) saturation contrast for 100 nM WT MBP and 100 nM MBP(329Y)-GFP as a function of percent MBP in maltose-bound closed conformation. For WT MBP, [maltose] = 0, 0.1, 0.3, 0.5, 1, 3, 1000 \(\mu M\). For MBP(329Y)-GFP, [maltose] = 0, 32, 72, 140, 5000 nM. Pulse length, \(\tau_{\text{pulse}} = 1.0496\) ms; field strength, \(B_{1,\text{max}} = 279\) \(\mu T\). The number of pulses increased linearly from 0 to 15 000.
between on- and off-resonance saturation transfer was measured as on-resonance hyper-CEST contrast. By this method, 100 nM MBP with 1 mM maltose reported 0.26 ± 0.01 saturation contrast; by comparison, 0.23 ± 0.02 saturation contrast was observed for 100 nM Bla.\(^1\) To evaluate the responsiveness of MBP MR contrast to maltose, additional saturation contrast measurements were taken for 100 nM MBP with decreasing maltose. The lowest maltose concentration that could be readily detected by MBP was 100 nM, with an observed saturation contrast of 0.050 ± 0.007. For each maltose concentration, the amount of $\text{MBP}_{\text{closed}}$ (i.e., contrast “ON”) was calculated using a $K_d$ of 1 µM. The observed saturation contrast was linearly proportional ($R^2 = 0.953$) to the percentage of MBP in the maltose-bound closed conformation, which provides a measure of maltose concentration (Fig. 2b).

**Mutagenesis to modulate maltose detection sensitivity**

To extend the detection threshold of MBP, Ile-329 was mutated to Tyr to greatly increase maltose affinity ($K_d = 22$ nM).\(^4^1\) This mutation alters the conformational dynamics of MBP to disfavor the unliganded, open conformation, thereby promoting maltose binding through conformational coupling.\(^4^1,4^2\) Importantly, Ile-329 is located in a “hinge region” opposite the maltose-binding cleft, far enough away from the Xe-binding site to reasonably assume that mutations at this position should not affect Xe exchange. The I329Y mutation was introduced to a MBP–GFP fusion construct prepared to facilitate protein quantitation during cell studies (vide infra). The Xe hyper-CEST z-spectrum of MBP(I329Y)–GFP in the presence of 1 mM maltose is comparable to WT MBP (Fig. S3†), though the saturation response of Xe@MBP(I329Y)–GFP is shifted 5 ppm downfield and is slightly attenuated compared to WT. Saturation contrast measurements for 100 nM MBP(I329Y)–GFP followed the same procedure used for WT MBP, but with the saturation frequencies of d-SNOB pulses positioned +100 ppm and −100 ppm (Fig. S4†). As observed for WT MBP, saturation contrast was linearly proportional ($R^2 = 0.997$) to the percentage of MBP in the maltose-bound closed conformation (Fig. 2b). The detection threshold for the I329Y mutant was 32 nM maltose, which gave rise to saturation contrast of 0.07 ± 0.01. These saturation contrast data demonstrate that MBP can be “tuned” through mutagenesis to detect maltose across varying concentration ranges. High-affinity mutants such as I329Y can be employed for nM-to-low µM maltose detection, whereas WT MBP can be used for low µM-to-mM maltose detection.

**Hyper-CEST of MBP in *E. coli***

To evaluate the hyper-CEST contrast detectable from MBP in a cellular environment, MBP with a C-terminal GFP tag (MBP–GFP) was expressed in BL21 (DE3) *E. coli* cells and time-dependent saturation transfer measurements were taken following the same protocol used for purified WT MBP (Fig. S5†). Using the GFP tag, MBP concentration was quantified in real-time by fluorescence intensity at 510 nm ($\lambda_{ex} = 489$ nm). *E. coli* growths not induced with IPTG served as controls to measure background contrast. Cells were washed with PBS and then transferred to an NMR tube for data collection. Control *E. coli* reported saturation contrasts of 0.11 ± 0.01 and 0.09 ± 0.01 with and without 1 mM maltose, respectively. Background contrast at 95 ppm downfield from Xe(aq) likely arises from Xe exchanging with the hydrophobic interior of cellular membranes. *E. coli* expressing MBP in the presence of maltose reported 0.25 ± 0.02 saturation contrast (Table 1), nearly five-fold higher than *E. coli* expressing MBP in the absence of maltose (0.14 ± 0.01), after subtraction of background (0.11 ± 0.01) from both. This highlights a mechanism for designing xenon-based MRI molecular imaging agents capable of detecting a specific analyte in cellular milieu.

**Multiplexed MR contrast**

Table 1. Hyper-CEST data for MBP in *E. coli*

| *E. coli* sample | $T_{1on}$ (s) | $T_{1off}$ (s) | Saturation contrast | [MBP–GFP]\(^a\) [µM] |
|------------------|--------------|--------------|-------------------|------------------------|
| Non-induced, no maltose | 19.1 ± 0.8 | 26 ± 2 | 0.09 ± 0.01 | <0.001 |
| Non-induced, 1 mM maltose | 18.9 ± 0.8 | 28 ± 2 | 0.11 ± 0.01 | <0.001 |
| Induced, no maltose | 17.9 ± 0.7 | 28.5 ± 0.8 | 0.14 ± 0.01 | 1 |
| Induced, 1 mM maltose | 13.3 ± 0.6 | 8 ± 2 | 0.25 ± 0.02 | 1 |

\(^a\) MBP–GFP concentration measured by fluorescence.
CEST contrast dependence on MBP conformation

To ascertain whether hyper-CEST contrast with MBP results from the closed conformation induced by maltose binding or from sugar binding alone, the z-spectrum of MBP in the presence of β-cyclodextrin (βCD) was obtained (Fig. S6†). βCD binds MBP with good affinity ($K_d = 1.8$ μM) in the same cleft as maltose, but its larger size prevents MBP from adopting a closed conformation.46 The z-spectrum of MBP with βCD shows no downfield saturation response, indicating that the maltose-bound closed conformation is required for producing hyper-CEST contrast. The lack of saturation contrast with MBP + βCD was unexpected given that Xe has higher affinity for MBP + βCD than MBP + maltose. Previous $^1$H–$^{15}$N HSQC NMR experiments measured a $K_d$ of 20 $\pm$ 10 M$^{-1}$ for Xe binding to MBP + βCD and concluded that $K_d$ was too low to measure for Xe binding to MBP + maltose: the addition of Xe to MBP + maltose produced no measurable changes in the $^1$H–$^{15}$N HSQC resonances.44 Indeed, the fact that the Xe(aq) peak is broadened more by βCD (FWHM = 60 ppm) than by maltose (47 ppm) reveals a greater degree of Xe exchange with MBP + βCD than MBP + maltose (Table S1†). The structural basis for a difference in Xe affinity is not obvious, however, as the contours of the Xe-binding cavity are largely conserved among several MBP complexes (Fig. S7†). It therefore appears that the primary determinant of hyper-CEST contrast in MBP is rate of Xe exchange, not Xe affinity. Xe exchange is likely too fast (∼MHz frequency) with MBPopen (with or without βCD), but the binding of maltose and the conformational change to MBPclosed slows the rate of Xe exchange to resolve a separate peak in the z-spectrum.

Mutagenesis to identify origin of hyper-CEST contrast

We performed mutagenesis experiments to further probe the mechanism of Xe CEST with MBP. Val-293 was chosen as the site for single-point mutations due to its proximity to bound Xe identified in the maltose-free MBP structure (Fig. 1). The methyl carbons of Val-293 are only 4.0 and 4.2 Å from Xe, respectively, thus it was expected that mutations at this position would affect Xe binding affinity and/or kinetics. Val-293 was mutated to Leu to reduce the cavity volume, and to Ala to enlarge the cavity. CD and fluorescence spectroscopy, along with thermal stability assays, confirmed that the V293L and V293A mutants retained the same global structure and maltose-binding properties as WT MBP (Fig. S8 and S9, Table S2†).

Hyper-CEST z-spectra of V293L and V293A were obtained following the same protocol used for WT MBP (Fig. S10†). V293L with and without maltose showed only a saturation response for $^{129}$Xe(aq), suggesting that substitution of valine for the bulkier leucine sidechain effectively blocks Xe from occupying the MBP cavity. This result helps to confirm that the crystallographically determined leucine site for MBPopen is also the site of hyper-CEST with MBPclosed. The $^{129}$Xe(aq) peaks for V293L are narrower (19.1 ppm without maltose, 21.5 ppm with maltose) than WT MBP (41.8 ppm without maltose, 47 ppm with maltose) (Table S1†), further confirming the overall reduction of Xe exchange with V293L relative to WT MBP. However, the $^{129}$Xe(aq) Peaks for V293L are broader than PBS in the absence of protein, likely due to non-specific Xe–protein interactions elsewhere on the surface of the protein.47

The z-spectrum of 80 μM V293A without maltose showed only the $^{129}$Xe(aq) peak, though in the presence of maltose a plateau of saturation response was observed between 50 and −10 ppm. Lowering the V293A concentration to 10 μM resolved this broad saturation response into two peaks—one at 0 ppm for $^{129}$Xe(aq), the other at 36 ppm for Xe@V293Aclosed. This large change in chemical shift from 95 ppm observed for WT follows a trend observed in T4 lysozyme,31 clathrate cages,48 and zeolites,49 where it has been noted that larger cavities produce smaller downfield $^{129}$Xe chemical shifts, and vice versa. Notably, this mutation increased the magnitude of CEST saturation contrast with 100 nM MBP from 0.26 ± 0.01 (WT) to 0.35 ± 0.02 for V293A (Fig. S11†). The molecular features of this signal enhancement and chemical shift change are under investigation.

Conclusions

In summary, we have demonstrated that 100 nM MBP generates significant saturation contrast in vitro and that observed contrast is proportional to % MBPclosed, thereby characterizing MBP as a “smart” analyte-sensitive biosensor. The I329Y MBP sensor at 100 nM concentration detected maltose in the range 32 nM to 5 μM, whereas 100 nM WT MBP detected maltose in the range 100 nM to 1 mM. Notably, the large (+95 ppm) $^{129}$Xe NMR chemical shift was generated within the GE MBP molecule and did not require post-translational modification or cofactor such as a lanthanide or other paramagnetic shift agent. Additionally, WT MBP appended with GFP was readily detected at 1 μM via hyper-CEST NMR when expressed in E. coli. The large downfield shift of WT MBP (δ = 95 ppm) makes it compatible with Bla (δ = 60 ppm) for multiplexing applications and ratiometric analysis. Notably, Xe–Blu was cleanly detected in the multiplexing experiment with minimal crosstalk from Xe–MBP.
by withholding maltose until Xe-MBP signal was desired. Experiments with 1329Y MBP confirm 3–4 orders-of-magnitude higher small-molecule sensitivity than that achieved with available GE T1 or 1H-CEST contrast agents. Indeed, nM maltose detection via MBP hyper-CEST NMR rivals the small-molecule detection sensitivity of many GE fluorescent sensors.

The increased contrast generated by V293A (and loss of contrast observed for V293L) confirms that MBP-CEST efficiency can be enhanced with mutations to the xenon binding site: coupling this with mutations promoting the maltose-bound, MBP-closed conformation\textsuperscript{5,6} should yield a superior biosensor. Furthermore, the modulated saturation frequency of V293A (δ = 36 ppm, shifted 59 ppm upfield from WT MBP) suggests that rational mutagenesis will yield MBP variants with a broad range of \(^{129}\text{Xe}\) NMR chemical shifts, akin to the palette of fluorescent proteins such as GFP and mCherry commonly used for multiplexed cellular imaging.\textsuperscript{28} Similar attempts to engineer Bla through site-directed mutagenesis have so far failed to improve or modulate its CEST signal (unpublished data), which makes MBP a particularly versatile protein system for elucidating the hyper-CEST mechanism. Finally, MBP variants that have been engineered previously to bind different ligands\textsuperscript{41,42} highlight the exciting potential for employing MBP-enhanced \(^{129}\text{Xe}\) NMR/MRI to detect bioactive molecules present in mammalian cells.

**Conflicts of interest**

There are no conflicts of interest to declare.

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**Notes and references**

1. G.-L. Davies, I. Kramberger and J. J. Davis, *Chem. Commun.*, 2013, 49, 9704–9721.
2. A. C. Esqueda, J. A. López, G. Andreu-de-Riquer, J. C. Alvarado-Monzón, J. Ratnakar, A. J. M. Lubag, A. D. Sherry and L. M. De León-Rodriguez, *J. Am. Chem. Soc.*, 2009, 131, 11387–11391.
3. S. Zhang, R. Trokowski and A. D. Sherry, *J. Am. Chem. Soc.*, 2003, 125, 15288–15289.
4. A. Y. Louie, M. M. Hüber, E. T. Ahrens, U. Rothbächer, R. Moats, R. E. Jacobs, S. E. Fraser and T. J. Meade, *Nat. Biotechnol.*, 2000, 18, 321–325.
5. G. Angelovski and É. Tóth, *Chem. Soc. Rev.*, 2017, 46, 324–336.
6. M. G. Shapiro, G. G. Westmeyer, P. A. Romero, J. O. Szablowski, B. Küster, A. Shah, C. R. Otey, R. Langer, F. H. Arnold and A. Jasanoff, *Nat. Biotechnol.*, 2010, 28, 264–270.
7. M. T. McMahon and A. A. Gilad, *Top. Magn. Reson. Imag.*., 2016, 25, 197–204.
8. N. Oskolkov, A. Bar-Shir, K. W. Y. Chan, X. Song, P. C. M. van Zijl, J. W. M. Bulte, A. A. Gilad and M. T. McMahon, *ACS Macro Lett.*, 2015, 4, 34–38.
9. G. Angelovski, *Angew. Chem., Int. Ed.*, 2016, 55, 7038–7046.
10. M. S. Albert, G. D. Cates, B. Driehuys, W. Happer, B. Saam, C. S. Springer and A. Wishnia, *Nature*, 1994, 370, 199–201.
11. P. Berthault, G. Huber and H. Desvaux, *Prog. Nucl. Magn. Reson. Spectros.*, 2009, 55, 35–60.
12. M. G. Mortuza, S. Anala, G. E. Pavlovskaya, T. J. Dieken and T. Meersmann, *J. Chem. Phys.*, 2003, 118, 1581–1584.
13. D. A. Barskij, A. M. Coffey, P. Nikolau, D. M. Mikhailov, B. M. Goodson, R. T. Branca, G. J. Lu, M. G. Shapiro, V. Telkki, V. V. Zhivonitko, I. V. Koptyug, O. G. Salnikov, K. V. Kvtunov, V. I. Bukhtiyarov, M. S. Rosen, M. J. Barlow, S. Safavi, I. P. Hall, L. Schröder and E. Y. Chekmenev, *Chemistry*, 2017, 23, 725–751.
14. Y. Wang and I. J. Dmochowski, *Acc. Chem. Res.*, 2016, 49, 2179–2187.
15. M. M. Spence, S. M. Rubin, I. E. Dimitrov, E. J. Ruiz, D. E. Wemmer, A. Pines, S. Q. Yao, F. Tian and P. G. Schultz, *Proc. Natl. Acad. Sci. U. S. A.*, 2001, 98, 10654–10657.
16. Q. Wei, G. K. Seward, P. A. Hill, B. Patton, I. E. Dimitrov, N. N. Kuzma and I. J. Dmochowski, *J. Am. Chem. Soc.*, 2006, 128, 13274–13283.
17. J. M. Chambers, P. A. Hill, J. A. Aaron, Z. Han, D. W. Christianson, N. N. Kuzma and I. J. Dmochowski, *J. Am. Chem. Soc.*, 2009, 131, 563–569.
18. Y. Wang, B. W. Roose, J. P. Philbin, J. L. Doman and I. J. Dmochowski, *Angew. Chem., Int. Ed.*, 2016, 55, 1733–1736.
19. C. Witte, V. Martos, H. M. Rose, S. Reinke, S. Klippel, L. Schröder and C. P. R. Hackenberger, *Angew. Chem., Int. Ed.*, 2015, 54, 2806–2810.
20. V. Roy, T. Brotin, J.-P. Dutasta, M.-H. Charles, T. Delair, F. Mallet, G. Huber, H. Desvaux, Y. Boulard and P. Berthault, *ChemPhysChem*, 2007, 8, 2082–2085.
21. K. Jeong, C. C. Slack, C. C. Vassiliou, P. Dao, M. D. Gomes, D. J. Kennedy, A. E. Truxal, L. J. Sperling, M. B. Francis, D. E. Wemmer and A. Pines, *ChemPhysChem*, 2015, 16, 3573–3577.
22. S. Yang, W. Jiang, L. Ren, Y. Yuan, B. Zhang, Q. Luo, Q. Guo, L.-S. Bouchard, M. Liu and X. Zhou, *Anal. Chem.*, 2016, 88, 5835–5840.
23. B. A. Riggle, Y. Wang and I. J. Dmochowski, *J. Am. Chem. Soc.*, 2015, 137, 5542–5548.
24. L. Schröder, T. J. Lowery, C. Hilty, D. E. Wemmer and A. Pines, *Science*, 2006, 314, 446–449.
25. Y. Wang and I. J. Dmochowski, *Chem. Commun.*, 2015, 51, 8982–8985.
26. S. Korchak, W. Kilian, L. Schröder and L. Mitschang, *J. Magn. Reson.*, 2016, 265, 139–145.
27. Y. Bai, P. A. Hill and I. J. Dmochowski, *Anal. Chem.*, 2012, 84, 9935–9941.
28. Y. Bai, Y. Wang, M. Goulan, A. Driks and I. J. Dmochowski, *Chem. Sci.*, 2014, 5, 3197–3203.
29 M. G. Shapiro, R. M. Ramirez, L. J. Sperling, G. Sun, J. Sun, A. Pines, D. V. Schaffer and V. S. Bajaj, Nat. Chem., 2014, 6, 629–634.
30 R. F. Tilton and I. D. Kuntz, Biochemistry, 1982, 21, 6850–6857.
31 Y. Wang, B. W. Roose, E. J. Palovcak, V. Carnevale and I. J. Dmochowski, Angew. Chem., Int. Ed., 2016, 55, 8984–8987.
32 S. M. Rubin, M. M. Spence, I. E. Dimitrov, E. J. Ruiz, A. Pines and D. E. Wemmer, J. Am. Chem. Soc., 2001, 123, 8616–8617.
33 S. M. Rubin, S. Y. Lee, E. J. Ruiz, A. Pines and D. E. Wemmer, J. Mol. Biol., 2002, 322, 425–440.
34 W. Boos and H. Shuman, Microbiol. Mol. Biol. Rev., 1998, 62, 204–229.
35 J. Evenäs, V. Tugarinov, N. R. Skrynnikov, N. K. Goto, R. Muhammad and L. E. Kay, J. Mol. Biol., 2001, 309, 961–974.
36 I. L. Medintz and J. R. Deschamps, Curr. Opin. Biotechnol., 2006, 17, 17–27.
37 F. S. Grünewald, in Advances in Chemical Bioanalysis, ed. F.-M. Matysik, Springer, Cham, 2013, vol. 6, pp. 205–235.
38 J. S. Marvin, E. R. Schreiter, I. M. Echevarria and L. L. Looger, Proteins, 2011, 79, 3025–3036.
39 M. Fehr, W. B. Frommer and S. Lalonde, Proc. Natl. Acad. Sci. U. S. A., 2002, 99, 9846–9851.
40 D. E. Benson, D. W. Conrad, R. M. de Lorimier, S. A. Trammell and H. W. Hellinga, Science, 2001, 293, 1641–1644.
41 M.-H. Seo, J. Park, E. Kim, S. Hohng and H.-S. Kim, Nat. Commun., 2014, 5, 3724.
42 J. S. Marvin and H. W. Hellinga, Nat. Struct. Biol., 2001, 8, 795–798.
43 G. Guntas, T. J. Mansell, J. R. Kim and M. Ostermeier, Proc. Natl. Acad. Sci. U. S. A., 2005, 102, 11224–11229.
44 J. S. Marvin and H. W. Hellinga, Proc. Natl. Acad. Sci. U. S. A., 2001, 98, 4955–4960.
45 M. H. Lee, J. S. Kim and J. L. Sessler, Chem. Soc. Rev., 2015, 44, 4185–4191.
46 A. J. Sharff, L. E. Rodseth and F. A. Quiocio, Biochemistry, 1993, 32, 10553–10559.
47 S. M. Rubin, M. M. Spence, A. Pines and D. E. Wemmer, J. Magn. Reson., 2001, 152, 79–86.
48 J. A. Ripmeester, C. I. Ratcliffe and J. S. Tse, J. Chem. Soc., Faraday Trans. 1, 1988, 84, 3731–3745.
49 J.-L. L. Bonardet, J. Fraissard, A. Gedeon and M.-A. Springuel-Huet, Catal. Rev.: Sci. Eng., 1999, 41, 115–225.
50 N. C. Shaner, P. A. Steinbach and R. Y. Tsien, Nat. Methods, 2005, 2, 905–909.