Permeant Ion-dependent Changes in Gating of Kir2.1 Inward Rectifier Potassium Channels

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ABSTRACT We studied the effect of monovalent thallium ion (Tl+) on the gating of single Kir2.1 channels, which open and close spontaneously at a constant membrane potential. In cell-attached recordings of single-channel inward current, changing the external permeant ion from K+ to Tl+ decreases the mean open-time by ~20-fold. Furthermore, the channel resides predominantly at a subconductance level, which results from a slow decay (τ = 2.7 ms at −100 mV) from the fully open level immediately following channel opening. Mutation of a pore-lining cysteine (C169) to valine abolishes the slow decay and subconductance level, and single-channel recordings from channels formed by tandem tetramers containing one to three C169V mutant subunits indicate that Tl+ must interact with at least three C169 residues to induce these effects. However, the C169V mutation does not alter the single-channel closing kinetics of Tl+ current. These results suggest that Tl+ ions change the conformation of the ion conduction pathway during permeation and alter gating by two distinct mechanisms. First, they interact with the thiolate groups of C169 lining the cavity to induce conformational changes of the ion passageway, and thereby produce a slow decay of single-channel current and a dominant subconductance state. Second, they interact more strongly than K+ with the main chain carbonyl oxygens lining the selectivity filter to destabilize the open state of the channel and, thus, alter the open/close kinetics of gating. In addition to altering gating, Tl+ greatly diminishes Ba2+ block. The unblocking rate of Ba2+ is increased by >22-fold when the external permeant ion is switched from K+ to Tl+ regardless of the direction of Ba2+ exit. This effect cannot be explained solely by ion-ion interactions, but is consistent with the notion that Tl+ induces conformational changes in the selectivity filter.

KEY WORDS: permeation • Tl+ • conformational change • selectivity filter • backbone mutation

INTRODUCTION

In many ion channels, gating is tightly coupled to ion permeation and is strongly influenced by the nature of the permeant ion. For example, gating of the Cl− channel of Torpedo electroplax is coupled to the transmembrane electrochemical gradient of Cl− ions (Richard and Miller, 1990). Moreover, elevation of extracellular Cl− concentration increases the open probability and opening rate of GIC chloride channels, and Cl− movement within the pore appears to constitute the actual gating charge (Pusch et al., 1995; Chen and Miller, 1996). Likewise, permeant ions affect the open probability and gating of cardiac Na+ channels by binding to a site deep in the pore (Townsend et al., 1997; Townsend and Horn, 1999). In K+ channels, where the effect of permeant ions on gating has been studied most extensively, elevation of extracellular K+ concentration tends to increase channel open probability and the rate of activation, reduce the rate of closing (deactivation) and C-type inactivation, and accelerate the rate of recovery from C-type inactivation (Swenson and Armstrong, 1981; Clay, 1986; Matteson and Swenson, 1986; Neyton and Pelleschi, 1991; Demo and Yellen, 1992; Pardo et al., 1992; Lopez-Barneo et al., 1993; Gomez-Lagunas and Armstrong, 1994; Baukrowitz and Yellen, 1995, 1996; Levy and Deutsch, 1996a,b; Kiss and Korn, 1998; Wood and Korn, 2000). Addition of certain permeant ion species such as Rb+ and Cs+ that bind the pore tightly also impedes the closing of the channel (Swenson and Armstrong, 1981; Matteson and Swenson, 1986; Demo and Yellen, 1992). These effects have been generally explained by the “foot-in-the-door” model of gating, which proposes that the channel pore cannot close or undergo C-type inactivation with an ion in it.

The crystal structure of the prokaryotic KcsA channel shows that three permeant ions are bound in the pore (Doyle et al., 1998). Two ions reside in the selectivity filter and are coordinated by main chain carbonyl oxygens from the signature sequence, TXGYG (with X mainly being either V or I). The third ion is located in the water-filled central cavity, where it is stabilized in part by a negative electrostatic potential generated by the tilted pore helices (Doyle et al., 1998; Roux and MacKinnon, 1999). There is evidence that occupancy of the selectivity filter by permeant ions hinders C-type inactivation (Kiss and Korn, 1998; Kiss et al., 1999).
Inward rectifier K⁺ (Kir)* channels are gated by a variety of mechanisms, including intracellular Mg²⁺ ions and polyamines, G proteins, and ATP (Nichols and Lopatin, 1997). In addition, activated single Kir channels open and close spontaneously and exhibit a wide range of gating kinetics. In cloned G protein-coupled or ATP-sensitive Kir channels, single-channel gating is characterized by bursts of rapid openings (milliseconds time scale) separated by long closed intervals (Trapp et al., 1998; Enketchakul et al., 2000; Yakubovich et al., 2000). On the other hand, single Kir2.1 channels typically open for long periods of time (tens to hundreds of milliseconds) and exhibit three closed states (Choe et al., 1999; Lu et al., 2001). We recently have shown that the single-channel gating behavior of Kir2.1 is dramatically altered when the amide carbonyl oxygens of the two conserved glycines in the TXGYG signature sequence are changed to ester carbonyls, suggesting that conformational changes of the selectivity filter contribute directly to channel gating (Lu et al., 2001). To further test this hypothesis, we have examined in this study the effect of changing the permeant ion species from K⁺ to Tl⁺ on gating of single Kir2.1 channels. Tl⁺ has nearly the same radius as K⁺ (1.40 vs. 1.33 Å) and is highly permeable through K⁺ channels (Hille, 1992). Thus, it serves well as a substitute for K⁺. On the other hand, Tl⁺ has a higher polarizability and electron affinity than K⁺ (Lee, 1971; Marcus, 1997), therefore, it is expected to interact more strongly than K⁺ with polarizable ligand groups in the pore.

**Materials and Methods**

**Molecular Biology and Oocyte Expression**

C169V mutation was made by oligonucleotide insertion method and was confirmed by sequencing. Tandem tetramers containing varying numbers of C169V mutant subunits were constructed using a method described previously (Yang et al., 1995; Lu et al., 1999). Oocytes were obtained from adult Xenopus laevis and were injected with 0.2–1 ng of cRNA transcribed in vitro as described previously (Lu et al., 1999). Injected oocytes were maintained in an 18°C incubator.

**Electrophysiological Recordings**

Recordings were performed 1–7 d after RNA injection. Whole-cell current was measured by two-electrode voltage clamp (TEVC) with the oocyte clamp amplifier (model OC-725C; Warner Instruments). Recording glass electrodes were filled with 3 M KCl and had a resistance of 0.4–0.8 MΩ. The external solution contained either 100 mM KNO₃ or 100 mM TlNO₃, with 2 mM MgCl₂ and 1 mM HEPES (pH adjusted to 7.4 with NMDG). In some experiments, 2.5 mM EDTA was added to the solution. In anomalous mole fraction experiments, external solutions contained mixtures of different concentrations of KNO₃ or TlNO₃, with the total concentration of both salts being kept at 100 mM. Current was evoked from a holding potential of 0 or +10 mV either by a −100-mV voltage pulse or a voltage ramp from −100 to +80 mV with an increment of 0.5 mV/ms. In Ba²⁺ blocking experiments, the duration of the pulse or the interpulse interval was adjusted to achieve steady-state Ba²⁺ blockage and complete recovery from the blockage.

Single-channel recordings were obtained in cell-attached membrane patches. Electrodes were fabricated from Corning Pyrex glass tubes, coated with silgard, and heat polished to a resistance of 5–8 MΩ when filled with a solution containing 110 mM KCl, 10 mM HEPES, 9 mM EGTA, and 1 mM EDTA (pH adjusted to 7.4 with KOH). In some experiments, we also filled the pipette with a solution containing 120 mM KNO₃, 1 mM KCl, 10 mM HEPES, 9 mM EGTA, and 1 mM EDTA (pH adjusted to 7.4 with NMDG). No significant difference in single-channel properties was observed with either K⁺ solutions, and, therefore, the data were pooled together. When Tl⁺ was used as the permeant ion, the solution contained 120 mM TINO₃, 1 mM TiCl, 10 mM HEPES, 9 mM EGTA, and 1 mM EDTA (pH adjusted to 7.4 with NMDG). In Ba²⁺ blockage experiments, the pipette solution contained either 130 mM KCl or 130 mM TINO₃ and 1 mM TiCl, with 10 mM HEPES and a known concentration of BaCl₂ (pH adjusted to 7.4 with NMDG). All reagents were purchased from Sigma-Aldrich.

For TEVC recordings, data acquisition and analysis were performed with a customized program written in Axobasic (Axon Instruments Inc.). Single-channel current was recorded with the pClamp 6 program (Axon Instruments, Inc.) at a constant negative potential ranging from −40 to −200 mV. Unless indicated otherwise, currents were filtered at 1 kHz with an 8-pole Bessel filter, digitized at 2–10 kHz with a 12-bit A/D interface (Axon Instruments Inc.) and stored on a computer. All recordings were conducted at temperatures 21–23°C.

**Data Analysis**

The permeability ratios, \( P_{Tl}/P_{K} \), was calculated from changes of the reversal potential upon switching from K⁺ to Tl⁺ solution according to the equation: \[ E_{rev} = E_{rev, K} - \frac{RT}{F} \log \left( \frac{P_T}{P_K} \right), \] where \( R, T, \) and \( F \) have their usual thermodynamic meanings. The voltage dependence of Ba²⁺ blockage was determined using the Woodhull (1973) model according to the equation: \[ K_T(V) = K_0(0) \exp(\delta \Delta E/RT), \] where \( K_0(0) \) is the apparent dissociation constant at 0-mV membrane potential, \( \delta \) is the valence of the blocking ion, and \( \Delta E \) is the fractional electrical distance of the blocking ion binding site from the outside.

Single-channel current was analyzed with the pClamp6 program. Only patches that contained one channel were included in the analysis. Dwell-time distributions were constructed from records of varying length (from 2 to 30 min depending on the frequency of open/closed or blocked transitions) using logarithmic binning and square-root transformation and were fit with one or more exponential functions. No corrections were made for filter dead-time and missed events. The Ba²⁺ unblocking rate was computed as the inverse of the blocked time. Data are presented as mean ± SD (number of observations). Statistical analysis was done with \( t \) test.

**Results**

**Tl⁺ Is Highly Permeant in Kir2.1 Channels**

We first examined Tl⁺ permeation properties in Kir2.1 channels expressed in Xenopus oocytes. Fig. 1 A shows whole-cell currents evoked by voltage ramps from −100 to +60 mV, with either K⁺ or Tl⁺ as the external permeant ion. The currents displayed strong inward rectification with either species of ions. Upon switching the

*Abbreviations used in this paper: Kir channels, inward rectifier K⁺ channels; Kv channels, voltage-gated K⁺ channels; TEVC, two-electrode voltage clamp.
external solution from K\(^+\) to Tl\(^+\), the reversal potential was shifted to the positive direction (see close-up in Fig. 1 B) and the inward current increased significantly. The average shift of reversal potential is 12.6 \(\pm\) 3.6 mV, corresponding to a permeability ratio \(P_{\text{Tl}}/P_{\text{K}}\) of 1.65, which is similar to the value of 1.5 reported for a native inward rectifier K\(^+\) channel (Hagiwara and Takahashi, 1974). This result indicates that Tl\(^+\) is more permeant than K\(^+\) in Kir2.1 channels, as it is in other types of K\(^+\) channels (Hille, 1992). At a membrane potential of \(-100\) mV, the Tl\(^+\) current was 1.75-fold larger than the K\(^+\) current (Fig. 1 C). This increase appears to be mainly due to a reduced blockage of Tl\(^+\) current by external divalent cations—2 mM Mg\(^{2+}\) (which was added to reduce the leak current and maintain long-lasting stable recordings) as well as contaminating Ba\(^{2+}\)—since the Tl\(^+\) current was only 1.06 (\(\pm\)0.02, \(n = 5\))-fold larger than the K\(^+\) current when external divalent cations were chelated by EDTA.

Kir2.1 channels also displayed a strong anomalous mole fraction effect: when Tl\(^+\) and K\(^+\) ions were mixed together (while total concentration of both ions was kept at a constant level of \(100\) mM), the whole-cell current became smaller than with either ion alone outside (Fig. 1 C). This behavior is very similar to that observed in a native inward rectifier channel (Hagiwara et al., 1977), and is indicative of multiple ion binding sites and strong ion-ion interactions in the pore.

**Tl\(^+\) Alters Single-channel Gating Kinetics**

As we reported previously (Lu et al., 2001), Kir2.1 channels expressed in *Xenopus* oocytes undergo spontaneous open and close transitions when recorded in cell-attached patches at constant membrane potentials more negative than the potassium equilibrium potential. Fig. 2 A shows representative inward single-channel currents recorded at \(-100\) mV with K\(^+\) as the external permeant ion. The open- and closed-time distributions show that there are one open and three closed states (Fig. 2 C). The open-time was voltage-dependent, decreasing e-fold with 67-mV hyperpolarization, from 293 ms at \(-80\) mV to 45 ms at \(-200\) mV (Fig. 2 E). The two longer closed-times were also voltage-dependent, decreasing e-fold with 71- and 62-mV hyperpolarization, respectively (Fig. 2 F). The shortest closed-time was relatively independent of membrane voltage. Previous studies have shown that these closed states are not produced by open-pore blockage by external or internal blocking ions (Choe et al., 1999; Lu et al., 2001).

Single-channel behavior was greatly altered when Tl\(^+\) was used as the external permeant ion. Representative single-channel Tl\(^+\) currents recorded at \(-100\) mV from a cell-attached patch are shown in Fig. 2 B. Compared with K\(^+\) currents, two major changes in the gating kinetics are readily seen. First, the channel closed more frequently and the open-time was decreased by \(~20\) fold.
across the voltage range studied (Fig. 2, B and E), indicating a dramatic destabilization of the open state by Tl⁺. Second, although there were still three closed states (Fig. 2 D), the dwell-time of the two longer closed states were altered, and they remained relatively constant in the voltage range of −140 to −200 mV and increased rather than decreasing with hyperpolarization in the voltage range of −80 to −140 mV (Fig. 2 G). Both dwell times were shorter than those obtained with K⁺ as the permeant ion at −100 mV. On the other hand, as the membrane potential was hyperpolarized to −200 mV, both dwell times became longer than those obtained with K⁺ as the permeant ion. Due to these two effects, the overall open probability at −100 mV decreased from ~0.9 for K⁺ to ~0.6 for Tl⁺. These results indicate that the single-channel gating kinetics is strongly influenced by the species of the permeant ion.
Tl⁺ channel properties. We also recorded Tl⁺ levels were also observed (Fig. 2 B, arrowheads) but to a large conductance level, from which it relaxed to the closed level, from which it relaxed to the fully open level. When the channel opened from the closed level, it often jumped to a large conductance level, from which it relaxed to the fully open level. However, from the subconductance state, the channel predominantly enters a closed state and only occasionally transition back to the fully open level. This type of nonequilibrium gating is phenomenologically reminiscent of that seen in a Cl⁻ channel, which is coupled to Cl⁻ permeation (Richard and Miller, 1990). The observation that the decay of single-channel Tl⁺ current always occurs immediately after channel opening to the full level suggests that it is tightly coupled to Tl⁺ permeation. Since Tl⁺ interacts strongly with soft ligands, we hypothesized that the decay might be caused by interactions between Tl⁺ and the thiolate group of a pore-lining cysteine residue. C169 in the second transmembrane segment (TM2) appeared to be a perfect target. We previously have shown that the side chain of C169 reacts covalently with methanethiolate reagents applied internally and presumably faces the pore (Lu et al., 1999). The crystal structure of KcsA shows that the side chain of the analogous residue in that channel is at the same level as the cavity ion and points to the central cavity (Doyle et al., 1998). The single-channel current decay and the associated subconductance level were abolished when C169 was mutated to valine, its counterpart in Kir1.1 channels (Fig. 4 A), indicating that they were indeed produced by interactions between C169 and Tl⁺ ions permeating the pore. The mutant channel underwent transitions only between the closed and fully open levels, with an unitary current of 4.8 ± 0.1 pA (n = 4) at −100 mV.

Kir2.1 channels expressed in *Xenopus* oocytes are homotetramers and, thus, have four identical C169 thiol groups. Does Tl⁺ need to interact with multiple thiolate groups or with only one thiolate group to induce the decay and subconductance level? If the latter were the case, one would expect that a channel containing one C169 still exhibits the decay but with one-fourth of the frequency observed in the wild-type channel. To address this question, we constructed tandem tetramers containing one to three C169V mutant subunits by using a strategy described previously (Yang et al., 1995; Lu et al., 1999). Although single channels formed by tandem tetramers containing three C169V mutant subunits still showed the decay and subconductance level, those formed by tandem tetramers bearing one or two C169V mutant subunits (either adjacent or opposite to each other) did not (Fig. 5). These results indicate that Tl⁺ must interact with at least three C169 thiolate groups to induce the single-channel current decay and subconductance level.

**Figure 3.** Slow decay of single-channel Tl⁺ current. (A) 12 selected single-channel current events that exhibit a slow decay are aligned by their initial opening and are superimposed. The currents were recorded at −100 mV in a cell-attached membrane patch with Tl⁺ as the external permeant ion. (B) The dotted line shows the ensemble average of the traces in A. The decay phase is fitted by a single exponential function with a time constant of 2.7 ms (solid line).

**Tl⁺ Induces Decay of Single-channel Current and a Subconductance Level**

In addition to its dramatic effect on open/close gating kinetics, Tl⁺ also induced marked changes in sublevel behavior (Fig. 2 B). Notably, the channel resided primarily at a subconductance level (Fig. 2 B, arrows) rather than the closed or fully open level. When the channel opened from the closed level, it often jumped to a large conductance level, from which it relaxed to the predominant sublevel within several milliseconds. Fig. 3 A shows 12 superimposed single-channel current events with a slow decay, aligned by their opening to the full level. The ensemble average trace shows a decay phase that can be fitted by a single exponential function with a time constant of 2.7 ms (Fig. 3 B). The unitary current at −100 mV is 4.2 ± 0.2 pA for the fully open level and 3.1 ± 0.2 pA (n = 10) for the subconductance level. Transitions to other subconductance levels were also observed (Fig. 2 B, arrowheads) but were not investigated further due to their rare occurrence. We also recorded Tl⁺ single-channel current with filtering at 3 kHz and found no difference in single-channel properties.
Effect of Tl\(^{+}\)/H\(^{+}\) Ions on Kir2.1 Channel Gating Kinetics of Single-channel Tl\(^{+}\) Current Was not Altered by C169V Mutation

Although Tl\(^{+}\) did not induce decay of single-channel current and subconductance level in C169V mutant channels, it still greatly destabilized the open state (Fig. 4). The mean open time at \(-100\) mV was decreased from 87 ± 3 ms (\(n = 4\)) with K\(^{+}\) as the permeant ion to 10.8 ± 0.4 ms (\(n = 4\)) with Tl\(^{+}\) as the permeant ion, which is similar to that for the wild-type channel. With Tl\(^{+}\), the dwell time of the three closed states at \(-100\) mV was 0.63 ± 0.25, 5.1 ± 0.7, and 35 ± 6 ms (\(n = 4\)), respectively.

**Figure 4.** C169V mutation abolishes the slow decay of single-channel current and subconductance level. (A) Representative single-channel Tl\(^{+}\) currents recorded at \(-100\) mV from a C169V mutant channel in a cell-attached membrane patch. Notice the lack of decay of single-channel current and subconductance level. (B) Open- and closed-time distributions of channels recorded with Tl\(^{+}\) as the external permeant ion. The open-time distribution was best fit by a single exponential and the closed-time distributions by three exponentials. The time constants of the fits are similar to those for the wild-type channel (Fig. 2 D).

**Figure 5.** Tl\(^{+}\) must interact with three or four C169 thiolate groups to induce the slow decay of single-channel current and subconductance level. The traces show representative cell-attached single-channel Tl\(^{+}\) currents recorded at \(-100\) mV from four different types of channels. The channels are formed by tandem tetramers containing one to three C169V mutant subunits: WT-C169V-C169V-C169V (CVVV), WT-WT-C169V-C169V (CCVV), WT-C169V-WT-C169V (CVCV), and WT-C169V-WT-WT (CVCC).
respectively, which are also similar to those for the wild-type channel (0.7 ± 0.25, 4.5 ± 1.7, and 21 ± 11 ms, n = 10). These results suggest that the effect of Tl⁺ on single-channel gating kinetics is caused by interactions between Tl⁺ and other parts of the pore. The most likely region is the selectivity filter, where the more electron "hungry" Tl⁺ is expected to interact more strongly than K⁺ with the main chain carbonyl oxygens.

**Tl⁺ Diminishes Ba²⁺ Blockage**

To test the hypothesis that the effect of Tl⁺ on single-channel kinetics is exerted through its interactions with the selectivity filter, we investigated whether Tl⁺ induces conformational changes in the filter region. Such conformational changes, if there were any, would most likely be very subtle. To detect such conformational changes, we used Ba²⁺ as a probe and compared inhibition of K⁺ or Tl⁺ current by external Ba²⁺. Since Ba²⁺ binds to a high affinity site in the selectivity filter (Neyton and Miller, 1988a,b; Jiang and MacKinnon, 2000), its binding kinetics is expected to be sensitive to even very subtle conformational changes in the region.

In whole-cell two-electrode voltage clamp recordings, 30 µM Ba²⁺ had little effect on the peak K⁺ current evoked by a test pulse to −100 mV from a holding potential of 0 mV, but it produced a time-dependent blockage of the current during the voltage pulse (Fig. 6 A). As Ba²⁺ blocks K⁺ channels by one-to-one binding (Jiang and MacKinnon, 2000), the apparent Kd at a given membrane voltage can be calculated according to the Hill equation: Fractional I = 1/(1 + ([Ba²⁺]/Kd)), where fractional I is the ratio of steady-state current in the presence and absence of a given concentration of Ba²⁺. At −100 mV and with 30 µM Ba²⁺, the steady-state K⁺ current was reduced by 91 ± 5% (n = 22), yielding an apparent Kd of 2.97 ± 1.65 µM for Ba²⁺. By contrast, 3 mM Ba²⁺ blocked only 79 ± 5% (n = 10) of the Tl⁺ current (Fig. 6 B), giving rise to a 277-fold higher apparent Kd of 824 ± 229 µM. Thus, Ba²⁺ is much less effective in blocking Tl⁺ current than K⁺ current. This reduced blockage was also observed at other membrane voltages (Fig. 6 C). With either K⁺ or Tl⁺ as the permeant ion, the apparent Kd of Ba²⁺ decreased significantly with increasing negative membrane voltage between −40 and −120 mV, reaching a minimum at −140 mV, and then increased slightly at −160 mV (Fig. 6 C). At even more negative membrane voltages, steady-state Ba²⁺ blockage became markedly reduced, which is indicative of permeation of Ba²⁺ to the inside. The equivalent valence (z) derived from data points from −40 to −100 mV was identical (0.8) in the presence of K⁺ or Tl⁺, suggesting that in both cases...
Ba\(^{2+}\) blocks the channel by binding to a site similarly located in the membrane electrical field. This \(z\) value is similar to the value of 1.08 reported previously for Kir2.1 channels (Shieh et al., 1998).

A much reduced Ba\(^{2+}\) blockade of Tl\(^{+}\) current was also evident at the single-channel level. With K\(^+\) as the external permeant ion, 10 \(\mu\)M external Ba\(^{2+}\) induced frequent long-lived blocking events lasting 5.27 ± 1.03 s (\(n = 5\)) at −100 mV (Fig. 7 A). When Tl\(^{+}\) was used as the external permeant ion, however, a much higher concentration of Ba\(^{2+}\) (300 \(\mu\)M) was required to induce significant numbers of blocking events (Fig. 7 B). The blocked state had a much shorter mean lifetime of 112 ± 19 ms (\(n = 8\)) at −100 mV, corresponding to a 47-fold increase in the Ba\(^{2+}\) unblocking rate (\(k_{\text{off}}\)). The voltage dependence of Ba\(^{2+}\) \(k_{\text{off}}\) was mild with either K\(^+\) or Tl\(^{+}\) as the permeant ion (Fig. 8). Nevertheless, in both cases, the \(k_{\text{off}}\) decreased significantly with increasing negative membrane potential in the range between −60 and −120 mV (or −140 mV for Tl\(^{+}\)). With Tl\(^{+}\) as the permeant ion, the Ba\(^{2+}\) \(k_{\text{off}}\) at −60 and −140 mV was 11.4 ± 2.3 s\(^{-1}\) and 7.8 ± 1.8 s\(^{-1}\) (\(n = 7\)), respectively, and was significantly different from each other (\(P < 0.01\)). This is indicative that the bound Ba\(^{2+}\) exits mainly to the external side in this voltage range. When the membrane potential was increased beyond −120 mV (or −140 mV for Tl\(^{+}\)), the Ba\(^{2+}\) \(k_{\text{off}}\) started to increase again, indicating that the bound Ba\(^{2+}\) now leaves mainly to the internal side. However, regardless of the direction of Ba\(^{2+}\) exit, its dissociation rate was increased by >22-fold when the permeant ion was switched from K\(^+\) to Tl\(^{+}\).

Due to the long duration of the blocking events, we were unable to determine the Ba\(^{2+}\) blocking rate at the single-channel level. However, the apparent Ba\(^{2+}\) blocking rate can be obtained by using the equation \(k_{\text{un}} = k_{\text{off}}/K_{\text{d}}\), where \(k_{\text{off}}\) is obtained from single-channel recordings and \(K_{\text{d}}\) from whole-cell recordings. According to this method, the Ba\(^{2+}\) blocking rate at −100 mV is reduced by nearly sixfold when the external permeant ion is switched from K\(^+\) to Tl\(^{+}\).

Since C169 interacts with Tl\(^{+}\) and since the C169V mutation abolished Tl\(^{+}\)-induced decay of single-channel current and subconductance level, we asked whether it also eliminated Tl\(^{+}\)-induced changes in Ba\(^{2+}\) blockade. Ba\(^{2+}\) blocked whole-cell K\(^+\) or Tl\(^{+}\) current at −100 mV in C169V mutant channels with an apparent \(K_{\text{d}}\) of 2.2 ± 0.2 \(\mu\)M (\(n = 5\)) and 863 ± 405 \(\mu\)M (\(n = 8\)), respectively.

**Figure 7.** Blockage of single-channel K\(^+\) or Tl\(^{+}\) current by external Ba\(^{2+}\). (A and B) Single-channel K\(^+\) or Tl\(^{+}\) current were recorded at −100 mV from two separate cell-attached membrane patches with 10 \(\mu\)M (A) or 300 \(\mu\)M (B) Ba\(^{2+}\) in the pipette solution. The arrows in the single-channel traces and in the closed-time distributions indicate Ba\(^{2+}\) blocked states. The dashed lines indicate the zero current level. The current record in A was filtered at 100 Hz and digitized at 2 KHz.

**Figure 8.** Tl\(^{+}\) speeds up dissociation of Ba\(^{2+}\) from the channel. The Ba\(^{2+}\) unblocking rate determined from single-channel recordings is plotted against the membrane potential with either K\(^+\) or Tl\(^{+}\) as the external permeant ion (\(n = 3–9\)).
similar to that obtained in the wild-type channel. Furthermore, in single-channel recordings, the Ba\(^{2+}\) dwell time at \(-100\) mV was decreased from \(5.4 \pm 0.6\) ms \((n = 4)\) in the presence of external K\(^{+}\) to \(145 \pm 16\) ms \((n = 4)\) in the presence of external Tl\(^{+}\), again similar to that observed in the wild-type channel. These results indicate that the Tl\(^{+}\)-induced changes in Ba\(^{2+}\) blockage do not result from interactions between Tl\(^{+}\) and C169.

**DISCUSSION**

Permeation and gating are two fundamental biophysical properties of ion channels. Although these two processes were originally thought to be independent, there is increasing evidence that they are tightly coupled and that permeant ions strongly affect gating (see **INTRODUCTION**). In this study, we demonstrate that substitution of K\(^{+}\) with another permeant ion (Tl\(^{+}\)) dramatically alters gating of Kir2.1 channels in two ways. First, Tl\(^{+}\) induces a slow decay of the single-channel current, resulting in a dominant subconductance level. Second, Tl\(^{+}\) greatly destabilizes the open state of the channel and changes the open/close kinetics of gating.

**Slow Decay of Single-channel Current**

The slow decay of single-channel current immediately after channel opening (Figs. 2 and 3) is unusual and, to our knowledge, is unprecedented. Such decay was observed in every channel recorded with Tl\(^{+}\) (>30 patches with various recording conditions), but was never observed with K\(^{+}\) as the permeant ion (>100 patches including results from our previous studies), indicating that it is conferred by the unique chemical properties of Tl\(^{+}\). The observation that the C169V mutation eliminates the slow decay and the associated subconductance level indicates that they result from interactions between Tl\(^{+}\) and the thiolate group of C169. In KcsA, the residue (I100) occupying the same position as C169 points directly at the cavity ion (Doyle et al., 1998), and our previous study indicates that the thiol group of C169 is exposed to the cavity as it can be covalently modified by thiol reactive reagents applied to the internal side (Lu et al., 1999). Thus, the thiolate group of C169 is ideally situated to interact with the cavity ion. Since Tl\(^{+}\) has a higher electron affinity than K\(^{+}\) and tends to seek out, wherever possible, polarizable ligand groups (Lee, 1971; Taylor and Brothers, 1993), the interaction between Tl\(^{+}\) and the thiolate group is much stronger than that with K\(^{+}\). Indeed, covalent thiol derivatives of Tl\(^{+}\) (RSTl) have been synthesized (Lee, 1971). However, the reversibility of both the slow decay and subconductance level indicates that the interaction between Tl\(^{+}\) and the thiolate group of C169 is predominantly ionic.

How does such ionic interactions result in the slow decay of single-channel current and the subconductance level? One possibility is that each of the four thiolate groups binds one Tl\(^{+}\) independently, partially occluding the ion conduction pathway, and thereby reducing the single-channel current. Previous work has shown that covalent modification of one of the four subunits at a nearby position, 172, by a thiol reactive reagent does indeed produce partial inhibition of the single-channel current (Lu et al., 1999). Although such a partial blocking mechanism can provide an explanation for the Tl\(^{+}\)-induced subconductance level, it does not explain the slow decay. More important, our results indicate that Tl\(^{+}\) must interact with at least three thiolate groups simultaneously to induce the slow decay and subconductance level as mutant channels containing only one or two C169 thiol groups did not exhibit either behavior (Fig. 5).

An alternative possibility is that when Tl\(^{+}\) permeates the cavity, it initiates an interaction with at least three thiolate groups of C169. This process involves reorientation of and optimal coordination by the thiolate groups. Since side chain rotation is generally extremely rapid (on the time scale of \(10^{-11}-10^{-8}\) s; Creighton, 1993), it is unlikely to be a major cause of the slow rate of the decay, which is of millisecond time scale. Instead, the interaction between Tl\(^{+}\) and C169 most likely induces a slow conformational change of the ion conduction pathway (the exact location is unclear) that leads to a slow decrease of Tl\(^{+}\) flux. This conformational change occurs rarely in K\(^{+}\) but frequently in Tl\(^{+}\). However, it should be noted that in the open channel Tl\(^{+}\) ions are streaming through the pore at a rate of \(>2 \times 10^{7}\) ions/s at a membrane potential of \(-100\) mV. Hence, the residence time of each permeating ion in the cavity is <50 ns. This raises an intriguing question of how Tl\(^{+}\) is coordinated in the cavity by the thiolate groups of C169. Fig. 9 A presents two possible scenarios. In one case, one can imagine that as Tl\(^{+}\) ions flow through the cavity in single file, the thiolate groups constantly change their orientation and coordination to track the permeating Tl\(^{+}\) ions, triggering the slow conformational change. Although the interaction between the thiolate groups and each permeating Tl\(^{+}\) is very brief, the altered conformation is maintained as long as Tl\(^{+}\) permeation continues. In another scenario, a Tl\(^{+}\) may be tightly coordinated by three of the four thiolate groups, which induces and maintains the slow conformational change, but the cavity is sufficiently wide to permit Tl\(^{+}\) ions to flow by the bound Tl\(^{+}\). Previous work suggests that it indeed may be structurally and electrostatically possible to support non–single file conduction through the cavity of Kir2.1 channels (Lu et al., 1999). In that study, one or two monovalently charged moieties were covalently linked to the D172 position.
Effect of Tl\(^{+}\) Ions on Kir2.1 Channel Gating

and the partially modified channels were still able to conduct K\(^{+}\) current, albeit at a reduced rate.

**Fig. 9 A** depicts a model to qualitatively describe the slow decay of single-channel current and the subconductance level induced by Tl\(^{+}\). In this model, we postulate that, with Tl\(^{+}\) as the permeant ion, the channel can adopt a fully open state (O\(_{f}\)), a subconductance state (O\(_{s}\)), and two closed states, C\(_{f}\) and C\(_{s}\). We also assume that in C\(_{f}\) the thiol group of C169 does not interact with Tl\(^{+}\), either because the side chain orientation or ionization is not optimal or because the cavity is devoid of Tl\(^{+}\). With this assumption, C\(_{f}\)→C\(_{s}\) and C\(_{f}\)→O\(_{s}\) transitions are not permitted. The closed channel can open to O\(_{f}\) only from C\(_{f}\) (Fig. 9 C). This transition generates full amplitude Tl\(^{+}\) inward current. Permeating Tl\(^{+}\) then binds to at least one thiolate group of C169 and this binding triggers a slow conformational change of the ion conduction pathway, resulting in a slow decay of Tl\(^{+}\) current and a subconductance state O\(_{s}\). From O\(_{s}\), the channel can occasionally transition to O\(_{f}\) but most of the time it enters C\(_{s}\) (Fig. 9 C), which still retains the conformational change induced by Tl\(^{+}\). C\(_{s}\) can subsequently open to O\(_{f}\) or enter C\(_{f}\) (Fig. 9 C). The latter transition (C\(_{s}\)→C\(_{f}\)) involves a conformation change opposite that of the O\(_{f}\)→O\(_{s}\) transition and enables the channel to conduct full amplitude Tl\(^{+}\) current when it undergoes the C\(_{f}\)→C\(_{s}\) transition again.

According to this model, channel openings to the full level (C\(_{f}\)→O\(_{f}\) transitions) should be, in most cases, preceded by two closed states (i.e., C\(_{s}\) and C\(_{f}\)) and those to the subconductance level (C\(_{s}\)→O\(_{s}\) transitions) by only one closed state (i.e., C\(_{s}\); Fig. 9 C). We segregated these two types of openings and analyzed the closed-time distribution (at -100 mV) of their preceding closed states, and found that indeed full openings are preceded by two closed states with a dwell time of 6.7 ± 1.1 ms and 26 ± 6 ms (n = 3) and subconductance openings by a single closed state with a dwell time of 1.6 ± 0.2 ms (n = 3). These closed times are similar to those obtained with both types of openings mingled together (see RESULTS). The kinetics of the open→closed transitions are likely governed by interactions between permeant ions and the selectivity filter, as will be discussed below.
Permeant Ion-induced Conformational Changes in the Selectivity Filter

Our previous results with amide-to-ester mutations suggest that permeant ion-dependent conformational changes of the selectivity filter contribute directly to the spontaneous gating of single Kir2.1 channels (Lu et al., 2001). In this study we found that Tl$^+$ dramatically destabilizes the open state, resulting in much more frequent closures of the channel. This effect was not dependent on interactions between Tl$^+$ and C169 since it persisted in the C169V mutant channel. Instead, it most likely results from strong interactions between permeating Tl$^+$ ions and the main chain carbonyl oxygen atoms lining the selectivity filter.

The mechanism by which permeant ions produce channel closure has been studied by Choe et al. (1998) on Kir1.1b channels, which display only a brief closed state. These investigators found that the rate of entering the closed state varied with the extracellular K$^+$ concentration and was proportional to current amplitude, suggesting a linkage between channel closure and K$^+$ permeation (Choe et al., 1998). They proposed that K$^+$ ions occasionally trigger a conformational change of the pore during permeation, thus altering the permeation energy profile and turning themselves into blocking ions. It is possible that the three closed states in Kir2.1 channels also represent permeant ion-blocked states. Consistent with this hypothesis, the rate of entering the closed states is much higher with Tl$^+$ than with K$^+$ as the permeant ion (Fig. 2 E), probably owing to stronger interactions between Tl$^+$ and the selectivity filter. Furthermore, the dwell times of the closed states and their voltage dependence depend on the species of the permeant ion (Fig. 2, F and G), indicating that the closed channels can register which ions are trapped in the pore. The multiple closed states may represent occupation by different numbers of permeant-turned-blocking ions at different locations in the selectivity filter. The voltage dependence of the dwell time of the closed states (Fig. 2, F and G) further suggests that K$^+$ ions trapped in the closed channel can escape internally, whereas trapped Tl$^+$ ions cannot not permeate the channel.

In addition to the dramatic effects on gating, Tl$^+$ also greatly diminishes Ba$^{2+}$ blockage of the channel. In particular, the Ba$^{2+}$ unblocking rate is increased by $>$22-fold within a wide range of membrane potential when the permeant ion is switched from K$^+$ to Tl$^+$ (Fig. 8). What is the biophysical mechanism underlying this dramatic destabilization of Ba$^{2+}$ binding? X-ray crystallography shows that in KcsA Ba$^{2+}$ binds to a single site at the end of the selectivity filter close to the central cavity (Jiang and MacKinnon, 2000). Single-channel analysis of Ba$^{2+}$ blockage of Ca$^{2+}$-activated K$^+$ channels indicates that there are at least three ion binding sites in addition to the Ba$^{2+}$ binding site (Neyton and Miller, 1988a,b). Two sites are located external to the Ba$^{2+}$ site and are referred to as the external lock-in site and the enhancement site. One site is located internal to the Ba$^{2+}$ site and is termed the internal lock-in site. These sites have different affinities for different species of ions and their occupancy greatly affects the kinetics of Ba$^{2+}$ binding (Neyton and Miller, 1988a,b).

Assuming that the same set of ion binding sites is present in Kir2.1 channels, can the large difference in Ba$^{2+}$ unblocking rate with either K$^+$ or Tl$^+$ as the permeant ion be explained by the difference in the occupancy of these sites? In a Ba$^{2+}$-blocked channel, the internal lock-in site is in true equilibrium with the oocyte cytoplasm in the cell-attached recording configuration and, thus, is visited only by K$^+$ regardless of the species of the external permeant ion. On the other hand, since the external lock-in site has a micromolar affinity for Tl$^+$ and K$^+$ (Neyton and Miller, 1988a), it is always occupied by either species of ions. Therefore, the difference comes down to the occupancy of the enhancement site by Tl$^+$ or K$^+$. If Tl$^+$ has a higher occupancy of the enhancement site (or, indeed, of any site external to the Ba$^{2+}$ binding site) than K$^+$, one would expect that the voltage range of $-40$ to $-120$ mV, when the bound Ba$^{2+}$ exits mainly to the external side, the Ba$^{2+}$ unblocking rate would be decreased due to obstruction by Tl$^+$. This is contrary to our experimental results that shows a >22-fold increase when switching from K$^+$ to Tl$^+$ (Fig. 8). Thus, the effect of Tl$^+$ on Ba$^{2+}$ unblocking rate is difficult to explain based solely on electrostatic interactions between the ions. However, the effect is consistent with the idea that Tl$^+$ ions change the conformation of the selectivity filter as they permeate and interact with that pore region, and that such conformational changes diminish Ba$^{2+}$ binding. We propose that the same conformational changes also underlie the dramatic effect of Tl$^+$ on the kinetics of single-channel gating.

Permeant ion-dependent conformational changes have been reported for other types of K$^+$ channels. The C-type inactivation of voltage-gated K$^+$ (Kv) channels involves a conformational change of the outer mouth of the pore, including the selectivity filter (Yellen et al., 1994; Liu et al., 1996; Starkus et al., 1997; Yellen, 1998; Kiss et al., 1999). Elevation of extracellular [K$^+$], which increases the occupancy of the outer mouth of the pore by K$^+$, slows the rate of entry into the C-type inactivated state (Lopez-Barneo et al., 1993; Baukrowitz and Yellen, 1996; Kiss and Korn, 1998). On the other hand, manipulations that decrease the occupancy of the outer mouth of the pore speed up entry into the C-type inactivated state (Lopez-Barneo et al., 1993; Baukrowitz and Yellen, 1995, 1996). Another example of permeant ion-dependent conformational changes comes from studies on the modulation by K$^+$ of TEA blockage of Kv2.1...
channels (Immke et al., 1999; Immke and Korn, 2000; Wood and Korn, 2000). In this case, binding of K⁺ to a high-affinity site in the selectivity filter changes the structure of the outer mouth of the pore, resulting in an enhanced blockage of the channel by TEA.

This work was supported by grant HL58552 (to J. Yang) and fellowship NS11097 (to T. Lu) from the National Institutes of Health. J. Yang was a recipient of the McKnight Scholar Award.

Submitted: 12 July 2001
Revised: 24 August 2001
Accepted: 25 September 2001

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