INTRODUCTION

Waste landfills can be divided into sanitary and unsanitary landfills depending on the installation and operation of environmental facilities such as waterproof facilities for leachate and leachate treatment facilities. As for a sanitary landfill, the impact of the surrounding environment including underground water, especially, is not very significant. However, as for an unsanitary landfill, when it has been used or is being used, the introduction of leachate continuously generated will severely pollute the nearby underground water as well as the related surface water.

The Ministry of Environment divides the unsanitary landfill management into sorting and transfer techniques and stabilization techniques. Sorting and transfer techniques are to excavate unsanitary landfills whose usage has been terminated, select combustibles or subjects of economic value and reclaim again. As for these techniques, even when the pollution of underground water is ceased, the recovery of the surrounding underground water system exposed to contaminants for a long time will not take place. The currently highlighted onsite stabilization techniques have a problem that contaminants will still be introduced to the surrounding underground system even when the stabilization of a landfill is continued. Additionally, for unsanitary landfills, unplanned or design-absent construction leads to long-term generation of subsidence and pollution, resulting in great restrictions on the land use while the continued generation of bad smell and danger of explosion or fire necessitates a long period for sufficient stabilization, which is likely to lead to problems with early application. Also the small-scale and temporary reclamation makes it difficult to establish appropriate construction techniques and increases the construction costs. Hence, there is no methods for actual treatment techniques other than excavation, transfer and reclamation.

So this paper aims to provide a basic study on the reuse of a landfill based on assessment of the physiochemical characteristics and advance stabilization (BMP, AT4) of buried
wastes on Sindong landfill, Asan City, Chungcheongnam-do whose term of use is about to expire. It also aims to present strategic directions to cope with rapid paradigm changes of landfills whose terms of use are about to end in the civilized society.

**EXPERIMENTAL**

**Subject samples:** Samples were collected from upper and lower areas of the three sites on Sindong landfill, Asan City, Choongcheongnam-do, divided based on the length of burial time. The equipments used for collecting samples include excavators and Auger cranes. Analyses of physiochemical characteristics and biodegradability of samples collected by site and location were conducted.

**Analysis of physical components:** As the most important variable to assume the characteristics of buried wastes such as the types of buried wastes and calculation of caloric value, analyses of apparent density, physical composition and the three components covering water, combustibles and lime were conducted based on the standard wastes testing methods. The chemical component analysis covered chemical elements such as carbon, hydrogen, oxygen, sulfur and chlorine and the equipment used for this was the Automatic Elemental Analyzer.

**Theoretical estimation of gases generated and accumulated gases:** An elemental analysis of organic wastes that can be decomposed biologically was conducted to produce a chemical formula. And then, the following formula (Buswell’s equation) was applied with an assumption that all organic matters should be converted to carbon dioxide and methane to calculate the total amount of digestion gases that might be generated.\(^4\)\(^5\)

\[
\begin{align*}
C_\text{n}H_\text{a}O_{\text{b}}N\text{CS}_\text{d} + (n \cdot a/4 - b/2 + 7c/4 + d/2) H_\text{O} & \rightarrow \\
(n/2 - a/8 + b/4 - 3c/8 + d/4) CO_\text{2} & + \\
(n/2 + a/8 - b/4 - 4c/8 - d/4) CH_\text{4} + cNH_\text{3} + dH_\text{2}S
\end{align*}
\]

This formula covers the relations between chemical composition of organic matters and methane transfereference number. The data from the sample's elemental analysis was substituted to calculate the theoretical amount of gases that could be generated. As for organic matters with higher mole contents such as carbon and hydrogen, the methane transfereference number will be assessed relatively higher while oxygen and nitrogen in chemical composition of organic matters tends to be highly related to the carbon dioxide fraction. The contents of carbon, hydrogen, oxygen, sulfur and chlorine and the equipment used for this was the Automatic Elemental Analyzer.

**Nutrient medium and seeding sludge:** Microorganism media to be used in a biochemical methane potential (BMP) test were prepared according to the method of Shelton and Tiedje\(^6\). Firstly, 1 N NaOH was prepared and 40 g of NaOH was melted in distilled water. It was placed into a 1,000 mL volume flask until it reached the marked line. And then, 1 N HCl prepared - HCl 36.5 g was added in distilled water. It was placed into a 1,000 mL volume flask until it reached the marked line. The components and quantities used for the preparation of nutrient media are as presented in Table-1. The phosphate buffer solution and inorganic nutrients were diluted 10 times until the volume reached 1,000 mL while microelements were concentrated 1,000 times. The sample won’t be completely melted when making a medium. So while it is being mixed, the sample shall be collected with a pipette to make a medium. The phosphate buffer solution and 100 mL of inorganic nutrients were mixed with 1 mL of microelements. It was placed in a 1,000 mL volume flask until the marked line was reached along with distilled water. And then, a sterilization medium was placed in a sample vial. This was pressurized and sterilized for ten minutes in a sterilizer. As for the seeding sludge for injecting anaerobic microorganisms, the remainder of the material collected from an anaerobic bath at Asan Environment Affairs Agency and filtered several times.

### Table-1 Components and Quantities Used for the Preparation of Nutrient Media

| Compound          | Concentration (g/L) |
|-------------------|---------------------|
| Phosphate buffer  |                     |
| KH\(_2\)PO\(_4\)   | 0.27                |
| K\(_2\)HPO\(_4\)   | 0.35                |
| NH\(_4\)Cl         | 0.53                |
| Mineral salts      |                     |
| CaCl\(_2\)-2H\(_2\)O | 0.075               |
| MgCl\(_2\)-6H\(_2\)O | 0.1                 |
| FeCl\(_3\)-4H\(_2\)O | 0.02               |
| MnCl\(_2\)-4H\(_2\)O | 0.0005             |
| H\(_2\)BO\(_3\)    | 0.00005             |
| ZnCl\(_2\)         | 0.00005             |
| CuCl\(_2\)         | 0.00003             |
| NaMoO\(_4\)-2H\(_2\)O | 0.00001           |
| CoCl\(_2\)-6H\(_2\)O | 0.0005             |
| NiCl\(_2\)-6H\(_2\)O | 0.00005           |
| Na\(_2\)SeO\(_3\)  | 0.00005             |

**Assessment of wastes biodegradability (Anaerobic BMP TEST):** In the BMP test, a serum bottle was used to inject necessary anaerobic microorganisms and nutrients into a bottle. While the optimum conditions such as temperature and pH were maintained, the amount of gas generated and its composition were identified. Samples of buried wastes were collected from sites divided into w-1 (upper, lower), w-2 (upper, lower) and w-3 (upper, lower) (within 2 m for upper areas and 7 m for lower areas). They were dried at the temperature of 105 °C (autoclave) and exposed to coldness before reducing them to powder as small as 10mm or smaller.

In order to remove oxygen in a sample vial, nitrogen gas was injected to reach an anaerobic state and then microorganism media and sludges were injected at the rate of 1:10. First, 300 mL of media was placed before filling nitrogen gas and placing 30 mL of seeding sludges. Again, nitrogen gas was
filled before placing a substrate (1.5 g VS/L, sample). In order to make pH level 7 after filling nitrogen gas, pH was adjusted with 1 N NaOH and 1 N HCl before filling nitrogen gas again. And then, bicarbonate was placed and capping was applied to secure sufficient alkalinity, which is to suppress the reduction of pH. After capping, a three-way cock was used to place a glass syringe into a sample vial. The incubator temperature was set at 35 ºC and a sample vial was placed for culturing. As for the gas produced, the amount was checked with a glass syringe connected to a three-way cock. As for the gas analysis, a GC syringe was used to collect gases of about 50 µL and direct injection into GC (GC : Agilent 7890 A GC Chemstation) was conducted. The analytic conditions for GC are as presented in Table-2. The calibration curve was prepared on each analysis using 2.5, 5 and 10 % CH₄ standard gases.

**Wastes biodegradability assessment (aerotropic AT4 TEST):** The principle of AT4 is to measure the amount of oxygen reduced from pressure lowered based on the soda lime absorption of carbon dioxide generated from aerotropic microorganisms. Based on this principle, Oxitop Control OC110 produced by Germany’s WTW was used. The calculation was based on the respiratory quotients formula provided by WTW.

The collected samples were dried at 105 ºC and powered up to 10 mm or below. As for the 40 g of samples in an analytic bottle (1000 mL), water was saturated up to 50-70 % of the maximum water holding capacity (WHC). As an absorber, NaOH was placed and cultured in an incubator at 20 ºC. Carbon dioxide generated from aerotropic microorganisms for four days (96 h) was absorbed with NaOH to calculate the rate of oxygen consumption due to the pressure difference in a bottle.

**RESULTS AND DISCUSSION**

**Physical composition and the three components covering water, combustibles and lime:** The material covers the measurement of apparent density using a 20 L container for samples collected from three sites (upper and lower). Apparent density of wastes differ by compositional elements, moisture, forms and sizes of wastes as well as sources of generation and seasons.

Table-3 suggests that density is greater in lower areas of W-1, W-2 and W-3 while average apparent density is 0.53 kg/L. Table-4 shows physical composition. Among all contents of combustibles, the amount of vinyl and plastic materials was the highest, followed by paper materials. Plastic materials have the greatest impact of carbon contents in elemental composition. Table-5 shows that plastic materials have higher carbon contents as their decomposition is difficult despite a long period of time of burial. The contents of food are much lower than those in general wastes (20-30 %) while there are no food materials in the lower areas of W-2 and W-3. This means that decomposition of food is much faster than that of other organic matters. Also, contents of combustibles are very high, enabling us to assume that the contents of vinyl and plastic materials whose decomposition is slow in the properties of the existing wastes buried in Sa landfill. Table-6 shows the results from analyses of the three components covering water, combustibles and lime. The water content was 43.06 %, which is the highest result, in the lower area of W-1 while it was lowest at 25.38 % in the upper area of W-2. By area of sample collection, water contents were higher in lower areas than upper ones, which means that based on the characteristics of a sanitary landfill, there was an impact of shield skirts as leachate treatment facilities in the lower area of the landfill. The higher amount of combustibles in this landfill than in others may have come from the sample collection method. For this test, excavators and Auger cranes were used to collect samples. Hence, the wastes buried could be evenly collected. Still, using Auger cranes only is unlikely to properly collect wastes buried and a lot of cover materials will be accompanied. This part necessitates clear standards for sample collection.

### TABLE-2

| Component          | Analysis method                      |
|--------------------|--------------------------------------|
| Gas chromatography | Agilent 7890 A GC Chemstation        |
| Detector           | FID (Flame Ionization Detector)      |
| Columns            | HP Capillary Columns 30 m            |
| Temperature        | 250 ºC                               |
| Pressure           | 10.3 pa                              |
| Oven temperature   | 35 ºC                                |
| Quantity           | 50 µL                                |

### TABLE-3

|                      | W-1 (upper) | W-1 (lower) | W-2 (upper) | W-2 (lower) | W-3 (upper) | W-3 (lower) | Average |
|----------------------|-------------|-------------|-------------|-------------|-------------|-------------|---------|
| Bulk density (kg/L)  | 0.459       | 0.594       | 0.423       | 0.514       | 0.542       | 0.663       | 0.53    |

### TABLE-4

|                      | Food | Paper | Wood | Plastic | Rubber/Leather | Fiber | Metal | Glass | Sand | Others | Total |
|----------------------|------|-------|------|---------|----------------|-------|-------|-------|------|--------|-------|
| W-1 (upper)          | 2.24 | 15.12 | 1.44 | 40.03   | 0.97           | 2.64  | 1.02  | 4.72  | 26.73| 5.10   | 100   |
| W-1 (lower)          | 0.51 | 7.42  | 1.52 | 40.40   | 0.88           | 7.55  | 4.67  | 5.62  | 26.78| 4.64   | 100   |
| W-2 (upper)          | 0.57 | 13.89 | 1.74 | 38.26   | 0.61           | 6.39  | 3.73  | 3.36  | 27.41| 4.04   | 100   |
| W-2 (lower)          | 0     | 0.74  | 1.14 | 45.22   | 0.87           | 16.42 | 9.34  | 3.32  | 18.33| 4.62   | 100   |
| W-3 (upper)          | 2.15 | 5.48  | 0.76 | 24.15   | 0.18           | 0.33  | 1.07  | 2.22  | 46.93| 16.75  | 100   |
| W-3 (lower)          | 0     | 1.31  | 2.45 | 19.54   | 0.83           | 4.33  | 3.72  | 1.16  | 49.44| 17.20  | 100   |
| Average              | 0.91 | 7.33  | 1.51 | 34.60   | 0.72           | 6.28  | 3.92  | 3.40  | 32.60| 8.72   | 100   |
Conditions. The oxygen consumption of buried wastes was 5.44–19.12 mg O₂/g dry matter while that in the area below W-1 was 19.12 mg O₂/g dry matter, which is the highest amount among the results. This may come from, just like in a BMP test, the impact of sample homogeneity and contents of organic materials. The result in the upper area of W-1 was 10.25 mg O₂/g dry matter. The result may have come from the reduced upper ground temperature in the winter and lowered activities of microorganisms. As for the measurement of the remaining three sites, the lower area of W-2 was 16.88 mg O₂/g dry matter and the upper area of W-3 was 5.4486 mg O₂/g dry matter while the lower area of W-3 was 7.6921 mg O₂/g dry matter. All the sample collection sites resulted in 25 mg O₂/g dry matter or below, the standard for assessment of advance stabilization.

As for the overall trends in data, the oxygen consumption of buried wastes with shorter periods of burial is higher while that in lower areas, rather than upper ones, resulted in higher oxygen consumption. This may have come from factors such as the reduction in the ground surface temperature and activities of microorganisms in the winter, contents of organic matters and homogeneity.

Conclusion

This paper presented a basic study on the reuse of waste landfills as well as the directions for reusing landfills based on analyses of the physical and chemical characteristics of wastes and biodegradability assessment. The test produced the following analytic results.

(1) Based on the results from analyses of physico-chemical characteristics, the contents of combustibles (about 60%) were very high compared to those of the existing landfills (about 30%). Based on the results from analyses of the three components covering water, combustibles and lime, water contents were higher in lower, than upper, areas on each site. This may have come from the characteristics of a sanitary landfill while the decrease of combustibles and the increase of lime can be attributed to decomposition of organic matters via microorganisms.

Result from wastes biodegradability analyses

Biochemical methane potential test: Table-7 shows that biodegradability of buried wastes is as low as 20% overall. Especially, as for the upper area of W-2, biodegradability was 4.41%. This may have come from much decomposition of organic materials due to reclamation. Fig. 1 shows the generation of methane gas, which means that organic matters of buried wastes have not yet been stabilized. This means that stabilization techniques for the reuse of a landfill shall take place in advance. Generally, when biodegradability is 5-7% or below, decomposition of organic matters can be understood to have been completed but there are no clear standards.

**TABLE-5**

| Chemical composition (weight ratio, %) |
|---------------------------------------|
| C | H | O | N | S | Cl | Ash |
|---|---|---|---|---|----|----|
| W-1 (upper) | 28.73 | 2.77 | 13.88 | 0.78 | 0.12 | 0.45 | 53.27 |
| W-1 (lower) | 21.04 | 3.07 | 22.15 | 0.72 | 0.02 | 0.60 | 52.40 |
| W-2 (upper) | 37.48 | 3.89 | 9.48 | 2.67 | 0.02 | 0.40 | 46.06 |
| W-2 (lower) | 9.86 | 1.53 | 7.38 | 0.34 | 0.03 | 1.15 | 79.71 |
| W-3 (upper) | 16.91 | 3.18 | 3.86 | 0.12 | 0.00 | 0.07 | 75.85 |
| W-3 (lower) | 14.34 | 2.36 | 4.90 | 0.22 | 0.01 | 0.14 | 78.04 |
| Average | 21.39 | 2.80 | 10.28 | 0.81 | 0.03 | 0.47 | 64.22 |

**TABLE-6**

| THREE COMPONENT ANALYSIS FOR EXCAVATED WASTE IN S LANDFILL (Unit: wt %) |
|-----------------------------|-------------|-------------|-------------|-------------|-------------|-------------|
| Moisture | Combustible | Ash | Total |
|---|---|---|---|
| W-1 (upper) | 30.68 | 34.97 | 34.35 | 100 |
| W-1 (lower) | 43.06 | 28.57 | 28.37 | 100 |
| W-2 (upper) | 25.38 | 50.65 | 23.97 | 100 |
| W-2 (lower) | 32.96 | 15.22 | 51.83 | 100 |
| W-3 (upper) | 27.07 | 29.42 | 43.51 | 100 |
| W-3 (lower) | 27.36 | 19.06 | 53.58 | 100 |
| Average | 31.09 | 29.65 | 39.27 | 100 |

**TABLE-7**

| THREE COMPONENT ANALYSIS FOR EXCAVATED WASTE IN S LANDFILL (Unit: wt %) |
|-----------------------------|-------------|-------------|-------------|-------------|-------------|-------------|
| Cumulative gas (mL CH₄/g VS) | 28.86 | 21.36 | 11.68 | 7.65 | 11.98 | 10.53 |
| Biodegradability (%) | 17.22 | 20.93 | 4.41 | 13.25 | 10.89 | 10.28 |

**TABLE-8**

| CUMULATIVE O₂ UPTAKE (AT4) OF WASTE IN S LANDFILL (Unit: mg O₂/g DM) |
|-----------------------------|-------------|-------------|-------------|-------------|-------------|-------------|
| W-1 (upper) | 10.25 | 19.12 | 11.43 | 16.88 | 5.44 | 7.69 |
| W-1 (lower) | 4.41 | 13.25 | 10.89 | 10.28 |     |     |
| W-2 (upper) | 7.69 | 22.15 | 13.63 | 15.43 | 10.43 | 10.28 |
| W-2 (lower) | 10.43 | 15.43 | 15.43 | 10.43 |     |     |
| W-3 (upper) | 10.53 | 10.53 | 10.53 | 10.53 |     |     |
| W-3 (lower) |     |     |     |     |     |     |

**Fig. 1.** Cumulative methane production for excavated W-1.
(2) Biodegradability of anaerobic (BMP TEST) buried wastes was about 20%. The generation of gases in samples stopped after about 10 days. Based on the results from aerotropic (AT4 TEST) assessment, oxygen consumption of buried wastes was 5.44-19.12 mg O₂/g dry matter. In all sample-collection area, the amount was 25 mg O₂/g dry matter or below, which is the standard for assessment of advance stabilization.

(3) As for more desirable treatment, combustibles on Sindong landfill, Asan City, Chungcheongnam-do can be dried and incinerated while non-combustibles can be re-buried. The selected landfill waste soils may be reused after their contents of organic foreign matters have been lowered than 1% via the secondary selection through washing or wind power.

This study helped us to secure a basic material on the reuse of Sindong landfill, Asan City, Chooncheongnam-do. This basic material will contribute to the future treatment and processing design for the reuse of Sindong landfill as well as establishing the methods to use and maintain landfills whose terms of use are about to end.

ACKNOWLEDGEMENTS

This work was supported by the Human Resources Development of the Korea Institute of Energy Technology Evaluation and Planning (KETEP) grant funded by the Korea Government Ministry of Knowledge Economy (No. 201140-10203130).

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