Relaxation Mechanisms in Glassy Dynamics: the Arrhenius and Fragile Regimes

H. George E. Hentschel\textsuperscript{1}, Smarajit Karmakar\textsuperscript{2}, Itamar Procaccia\textsuperscript{1} and Jacques Zylberg\textsuperscript{2}
\textsuperscript{1} Department of Chemical Physics, The Weizmann Institute of Science, Rehovot 76100, Israel
\textsuperscript{2} Departamento di Fisica, Universit\`a di Roma “La Sapienza”, Piazzale Aldo Moro 2, 00185, Roma, Italy
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Generic glass formers exhibit at least two characteristic changes in their relaxation behavior, first to an Arrhenius-type relaxation at some characteristic temperature, and then at a lower characteristic temperature to a super-Arrhenius (fragile) behavior. We address these transitions by studying the statistics of free energy barriers for different systems at different temperatures and space dimensions. We present a clear evidence for changes in the dynamical behavior at the transition to Arrhenius and then to a super-Arrhenius behavior. A simple model is presented, based on the idea of competition between single-particle and cooperative dynamics. We argue that Arrhenius behavior can take place as long as there is enough free volume for the completion of a simple $T1$ relaxation process. Once free volume is absent one needs a cooperative mechanism to ‘collect’ enough free volume. We show that this model captures all the qualitative behavior observed in simulations throughout the considered temperature range.

I. INTRODUCTION

A full understanding of the complex dynamical scenarios accompanying the glass transition requires a resolution of the statistics of free energy barriers at any given temperature. Such information is very hard to come by. Trying to measure such free energy barriers using the energy landscape of a typical glass former is a daunting procedure that has never been achieved. In this paper we propose a method that provides us with a decent approximation of the statistics of the free energy barriers. We do this by measuring “quenched” rather than “annealed” relaxation times. The usual procedure for extracting a relaxation time from numerical simulations of any glassy system is the “annealed” procedure. One takes many realizations of a super-cooled liquid at a given temperature, measures a typical relaxation function for each realization, averages the relaxation functions over the ensemble and finally extracts a relaxation time $\tau_\alpha(T)$ from this average relaxation function by, say, determining when the average function reaches $1/e$ of its initial value \cite{1}. The subscript $\alpha$ designates the ‘main’ slow relaxation mechanism in glassy dynamics, and is the only one that is typically resolved in numerical simulations. This procedure is automatically performed in experiments where the macroscopic nature of the relaxing systems provides self-averaging resulting in ‘smooth’ relaxation functions that hardly fluctuate from realization to realization. In this paper we propose that significant simulational insight on glassy relaxation can be gained by adopting a “quenched” procedure. In this procedure a relaxation time $\tau_\alpha$ is extracted from each and every realization by determining when the appropriate relaxation function reaches a value of $1/e$, cf. Fig. 1 upper panel. Finally an average is taken over the ensemble to provide $\langle \tau_\alpha(T) \rangle$. A typical such quenched relaxation time for a Lennard-Jones binary glass is shown in the lower panel of Fig. 1. Here we measured the self part of the intermediate scattering function for each realization separately (upper panel), i.e. the function

$$F_k(t;T) = \frac{2}{N} \sum_{i=1}^{N/2} \exp \{ i k \cdot |r_i(t) - r_i(0)| \}$$

(1)

where the index $i$ runs over half the particles in the binary mixture (those with longer interaction length; we assume the mass of all the particles to be the same, $m = 1$). One observes the usual dramatic slowing down, such that $\langle \tau_\alpha(T) \rangle$ grows rapidly when $T$ decreases, first in a pre-Arrhenius form (see inset), then in an Arrhenius form, linear in $1/T$, and later in a faster, super-Arrhenius form, which is referred to as ‘fragile’ behavior in the glass community \cite{2}. In this paper we will be interested predominantly in the transitions to and from the Arrhenius regime, for which the quenched procedure is particularly illuminating.

The advantage of the quenched over the annealed procedure is that it allows a particularly transparent treatment of the statistics of free energy barriers. Having an ensemble of $\tau_\alpha$ values for a given ensemble of super-cooled systems, we can define a typical free energy scale by writing for each $\tau_\alpha$ value an equation

$$\tau_\alpha = \tau_0 \exp \left( \frac{F}{T} \right) ,$$

(2)

where we choose units such that Boltzmann’s constant equals unity. Inverting Eq. (2), every value of $\tau_\alpha$ yields a value of $F$. The free energy has in principle a contribution from an energy barrier $E$ and an entropic contribution stemming from a degeneracy $g$,

$$F = E - T \ln g .$$

(3)

Of course, this definition makes sense when the dynamics involves escaping an energy barrier. The prefactor $\tau_0$ is by definition the relaxation time inside the confined state at every temperature $T$, and we assume that it is fixed for all the realizations in the ensemble:

$$\tau_0 = C/\sqrt{T} .$$

(4)
The dependence on $1/\sqrt{T}$ stems simply from the estimating $v$, the typical particle velocity, from equipartition as $v^2 = T$ and $\tau_0$ is then a typical length-scale (a cage magnitude) over $v$. For our purpose the constant $C$ is determined by requiring that the plot of $\tau_\alpha/\tau_0 \to 1$ as $T \to \infty$, see Fig. 1 (red cross) and is $O(1)$.

II. THE ARRHENIUS REGIME

Our first observation is that in the Arrhenius regime the distribution of free energies $F$ is always Gaussian, and is always very sharply peaked. Since $F$ in the Arrhenius regime is temperature independent, we conclude that $g = 1$, and $F = E$. As an example we show again the binary Lennard-Jones system for which we have used 1000 realizations for each value of $T$ to generate the distributions of $F$ shown in Fig. 2. We conclude that in the Arrhenius regime to a very good approximation

$$P(F) = \frac{1}{\sqrt{2\pi\sigma^2}} e^{-\frac{(F - \langle F \rangle)^2}{2\sigma^2}},$$

with the peak value $\langle F \rangle$ and the width $\sigma$ unchanged throughout the Arrhenius regime. In the present regime we denote $\langle F \rangle \equiv E_\alpha$. It is noteworthy that the distributions are also very sharply peaked; to see this note that the quenched average can be now computed from the Gaussian integral as

$$\langle \tau_\alpha \rangle(T) = \tau_0 \exp \left[ \frac{E_\alpha}{T} + \frac{\sigma^2}{T^2} \right].$$

Alternatively,

$$\ln \frac{\langle \tau_\alpha \rangle(T)}{\tau_0} = \frac{E_\alpha}{T} + \frac{\sigma^2}{T^2}.$$  

For the data shown in Fig. 2 $E_\alpha = O(1)$ and $\sigma \approx 0.05$. Thus for the Arrhenius range of temperatures $T = O(1)$.
the contribution of the width \( \sigma^2 \approx 2.5 \times 10^{-3} \) is of the order of 0.1\% in Eq. (4). Obviously, for larger systems we expect \( \sigma \) to be even smaller, tending to zero for \( N \to \infty \). It appears that all the relaxation events contributing to the Arrhenius regime belong to a tight group of similar events with a very well defined energy barrier \( E_a \) and a negligible dispersion. In fact, we find that this conclusion is not particular either to the Lennard-Jones system or to \( d = 2 \). In Fig. 4 we show similar results for other glass formers with A) purely repulsive potential \( \sigma \) at \( d = 2 \), B) the Lennard-Jones system \( \sigma \) at \( d = 3 \), C) the Kob-Andersen model \( \sigma \) at \( d = 2 \) and D) the polydisperse model \( \sigma \) at \( d = 2 \). The physical interpretation of these observations is discussed below, cf. Sect. VIII.

III. THE PRE-ARRHENIUS REGIME

At high temperatures we observe (see for example Fig. 3) that \( P(F) \) does not have a fixed dispersion (it widens when \( T \) increases) and its peak moves to lower free energy values. This observation is generic for all the models studied here. The physical interpretation of this observation is again discussed in Sect. VIII.

IV. TRANSITION TO SUPER-ARRHENIUS

Obviously, one might expect that when \( T \) reaches a value of \( T \approx \text{Const} \times E_a \) something interesting should happen. The physical relaxation mechanism whose energy barrier is \( E_a \) should run out of steam, and other mechanisms, if they exist, should emerge. This phenomenon is beautifully seen in the Lennard-Jones system, see Fig. 4. Indeed, at \( 1/T \approx 1.5 \) the Arrhenius process begins to disappear and we observe at first a gradual increase in both \( \langle F \rangle \) (the peak position) and its width \( \sigma \). In the system for which \( N = 6400 \), we find that at \( 1/T \approx 2.7 \) the high temperature Gaussian peak disappears in favor of a second Gaussian peak which appears to replace it. We will denote by \( T^* \) the temperature at which both process coexist with equal probability. The Gaussian of the second process marches to the right with increasing both of \( \langle F \rangle \) and \( \sigma \). The transition region is depicted in Fig. 4 and is shown in more detail in Fig. 5. For the systems of the size studied in these simulations, by the time that the second relaxation mechanism takes over, the contribution of the variance \( \sigma \) to the quenched average relaxation time Eq. (6) is no longer negligible. Thus for our example at \( T = 1/3 \langle F \rangle \approx 2 \) and \( \sigma \approx 0.25 \) such that \( \sigma^2/\langle F \rangle \approx 0.1 \). At lower temperature the dispersion effect will contribute more and more to the relaxation time. Of course, also here we expect \( \sigma \) to decrease to zero when \( N \to \infty \). It is useful however to keep the \( \sigma \) dependence for the simulations at hand.

We should note that Eq. (7) remains valid throughout the temperature range. We can rewrite it, denoting \( x_i \equiv \langle F_i \rangle /T \) and introducing \( w_i \) for the weighted sum of the two different mechanisms in the form

\[
\ln \left( \frac{\langle \tau \rangle(T)}{\tau_0} \right) = \sum_{i=1}^{2} w_i \left[ x_i + \frac{\sigma^2}{\langle F_i \rangle^2 x_i^2} \right].
\]

(8)

We test this formula for the data corresponding to the distributions found in Fig. 4 and demonstrate the perfect agreement in Fig. 6. Note that the relaxation time varies in this range of temperatures by close to 5 orders of magnitude.
different curves \[8, 9\]. Once plotted as a function of with an appropriate value of \(x_0\) collapsed by non-dimensionalizing the abscissa by the energy scale \(\varepsilon_r\) sets the energy scale of the interparticle potential, denoted as \(\varepsilon\). This energy scale depends on the system size, and there-

FIG. 6: Color online: A test of Eq. (7). Shown are data for the binary Lennard-Jones model at \(d=2\) and at different densities.

\[
\ln \frac{\langle \tau \rangle}{\tau_0} = \sum_{i=1}^{2} w_i \left[ \langle F_i \rangle / T + \sigma_i^2 / T^2 \right]
\]

FIG. 7: Color online. Upper panel: Usual plots of the non-dimensional relaxation time in logarithmic scale vs. \(1/T\) for five different values of the density, \(\rho = 0.80, 0.85, 0.90, 0.95, 1.00\). Lower panel: The very same data collapsed by non-dimensionalizing the abscissa by the energy scale \(\varepsilon \rho^y\).

\[
\ln \frac{\langle \tau \rangle}{\tau_0} \sim \varepsilon \rho^{y/d} / T
\]

\[
\langle F \rangle \sim \varepsilon \xi^{-y} \sim \varepsilon \rho^{y/d} .
\]

Thus by presenting \(\ln(\tau)/\tau_0\) vs. \(\varepsilon \rho^{y/d} / T\) we should find data collapse as discussed in Ref. \[7\]. It is well known that the argument remains approximately valid also when the potential is not a pure power law but with an exponent \(x\) that needs to be found by collapsing the data \[10\]. In the second panel of Fig. \[7\] we show the data collapse when plotted properly, with the density varying in the range \([0.80, 1.00]\). Additionally, in these systems the inter-particle potential possesses only one energy scale, which is nothing but the depth of the inter-particle potential, denoted as \(\varepsilon\). This energy scale must determine the typical barrier height that needs to be surmounted in the Arrhenius regime. Accordingly, we expect \(E_a\) to be proportional to \(\varepsilon\). This expectation is fully supported by the data shown in Fig. \[8\] where for two models (binary Lennard-Jones in 2 and 3 dimensions) we have changed \(\varepsilon\) and measured \(E_a\).

Our additional contribution to the discussion of density scaling is in pointing out that it is obeyed separately by the two contributions to Eq. (7). In the upper row of Fig. \[9\] we show that \(\langle F \rangle\) exhibits a very nice data collapse when data for different densities are re-plotted as required. In the lower row we show the same for \(\sigma\). We will use these scaling functions below in providing a model for the observed glassy dynamics.

V. SCALING

In Ref. \[7\] it was shown that relaxation times of models of the type discussed above exhibit density scaling. For different densities \(\rho\) plots of \(\ln(\langle \tau \rangle/\tau_0)\) vs. \(1/T\) fall on different curves \[8, 9\]. Once plotted as a function of \(\varepsilon \rho^{y/d} / T\) with an appropriate value of \(x\) the data collapses onto one curve. Density scaling is easiest to justify when one uses a purely repulsive potential of the form \(\phi(r) \sim \varepsilon r^{-y}\) where \(\varepsilon\) sets the energy scale of the interparticle potential and \(r\) is measured in dimensionless units. Then the only relevant length scale \(\xi\) in the system is determined by the density as \(\xi \sim \rho^{-1/d}\) where \(\rho\) is the density in dimensionless units. Obviously, when we plot \(\ln(\tau)/\tau_0\) vs. \(1/T\) (see upper panel of Fig. \[7\]) we present a di-

VI. SYSTEM SIZE DEPENDENCE OF \(T^*\)

An obvious worry about our method of analysis is that the dispersion \(\sigma\) depends on the system size, and therefore the temperature \(T^*\) at which we observe the transition to fragility may depend on the system size. This is of
VII. COOPERATIVE MECHANISM

The purpose of this section is to establish one of the main claims of this paper, i.e. that with the transition to the super-Arrhenius behavior one observes a change from a local to a cooperative relaxation mechanism. At temperatures below $T^*$ one expects a cooperative relaxation process to play an important role; the proposition of this paper is that this is mainly due to the disappearance of the free volume that was available in the Arrhenius regime (see quantitative model below), requiring now a cooperative motion of many particles to ‘collect’ enough free volume to allow a relaxation step. Although such a picture is current among some practitioners in the glass community, to the best of our knowledge it had never been shown explicitly. We use in this section scaling arguments to establish this picture convincingly.

Adding the system size dependence to our typical free energy, we write now

$$F(N, T) = E(T) - T \log[g(N, T)].$$  \hfill (11)

We expect $E(T)$ to be determined by the energy landscape and not to depend strongly on the system size. Energy barriers are mostly sensitive to local arrangements of particles which are not system size dependent. On the other hand $\log[g(N, T)]$ is the entropic contribution to the free energy barrier. Our data showed that in the Arrhenius regime the degeneracy factor $g$ was of the order of unity; here we will show that our data strongly supports a cooperative process in the super-Arrhenius regime, mainly due to the system size dependence of the degeneracy factor.

The flip side of having a cooperative mechanism is that there should exist a typical length $\xi(T)$ which measures the degree of cooperativity and is increasing when the temperature decreases. Consider then a system of $N$ particles in the super-Arrhenius regime, associated with a typical scale $\xi(T)$, and contained in a cubic box of size $L$ at some temperature $T$. As long as the system is small, i.e. $L \ll \xi(T)$, the whole system needs to cooperate in order to relax. Accordingly we expect the degeneracy factor to grow extensively like $g(N, T) \sim N$. When the system is large enough, i.e. when $L \gg \xi(T)$ we expect to find $g(N, T) \sim \xi^d$, not changing with the system size. In other words,

$$g(N, T) \sim \begin{cases} \left(\frac{\xi(T)}{a}\right)^d, & \text{for } L \ll \xi(T) \\ \left(\frac{\xi(T)}{a}\right)^d, & \text{for } L \gg \xi(T), \end{cases} \hfill (12)$$

where $a$ is typical inter particle distance. Accordingly the free energy should assume the form of a scaling function of $L/\xi(T)$ as

$$F(N, T) = E(T) - T \log \left[ A \left(\frac{\xi(T)}{a}\right)^d \left(\frac{L}{\xi(T)}\right)^f \right], \hfill (13)$$

course correct. Nevertheless we discover that the value of $T^*$ converges to a value $T^{**}$ in the thermodynamic limit. The data is shown for the binary Lennard-Jones model in Fig. [10]. From the data in the lower panel one can conclude that for this model

$$T^* - T^{**} \sim N^{-x}, \quad x \approx 0.25. \hfill (10)$$

We did not perform an exhaustive analysis of the exponents of this law of convergence for the different models.
where $A$ is a temperature independent proportionality factor. The scaling function $f(x)$ has the following asymptotic dependence

$$f(x) = \begin{cases} 
    x^d & \text{for } x \ll 1, \\
    1 & \text{for } x \gg 1.
\end{cases}$$

(14)

Considering the small system limit, we conclude that there the free energy barrier should exhibit a logarithmic dependence on the system size. This expectation is supported by the data shown in Fig 11. In the upper panel we show the system size dependence of the free energy value where the distribution function $P(F, T)$ has a second peak, for different temperatures. With increasing system size these values tend to saturate to an asymptotic value. Note that the dependence on the system size seems to be decreasing with increasing temperature, indicating the existence of a cooperative length scale. We can not go to high temperatures in this study because at higher temperature the extraction of the second peak position become difficult as its height goes down drastically, merging with the Arrhenius peak. We focus on data where the position of the second peak can be estimated accurately. To show the logarithmic dependence on the system size we collapsed the data together by rescaling the free energy axis by $\xi^d$ with $\delta \approx 1.2$ and the x axis by $\xi(T)^d$. For simplicity we fitted here a guess function for $\xi(T)$ in the form $\xi(T) \sim 1/T^{1.8}$ [11]. In the lower panel of Fig 11 one can see the convincing logarithmic dependence of the free energy on the system size for different temperatures.

VIII. INTERPRETATION AND DISCUSSION

The data shown above indicates two changes in the relaxation mechanism between the pre-Arrhenius and the Arrhenius regime and then between the Arrhenius and the fragile regime. We propose that the pre-Arrhenius regime is a fluid regime, in which a well-defined cage is only beginning to form around every particle. It is relatively easy to break this ill-formed cage, and there are many ways to do it. Accordingly the free energy needed is made of a low value of $E$ and a finite value of $g$ such that when $T$ is increasing the Gaussian peak that we observe marches to the left and becomes broader.

In the Arrhenius regime a well defined cage has formed around each particle, and this cage is a more or less regular cluster of particles aggregating around each center particle. We expect a roughly constant energy $E$ to suffice to break this kind of cage and the width of the distribution of $F$ should be relatively narrow as discussed below. Towards the end of the Arrhenius regime a second and maybe third layer of next-nearest neighbors begins to form around a center particle, and therefore it becomes more and more difficult for the single-particle relaxation to take place. The energy barrier increases but because of the fluidity of the second and third layer also the width of the distribution starts to increase.

Eventually a new, cooperative mechanism must be favored for dynamical relaxation. Note that the apparent mean energy of activation $\langle F \rangle$ increases when the temperature decreases below the Arrhenius regime. This must mean that the single particle relaxation mechanism that is operative at the Arrhenius regime is no longer available at lower temperatures, since if it were available it would have been selected. In the temperature range where both mechanisms coexist we indeed find the associated $\langle F \rangle$ and $\sigma$ to be similar. We thus offer a model of the observed phenomenology on the following basis:

(i) In the Arrhenius regime the relaxation process is an elementary event like a $T_1$ process which is characterized by a relatively sharp free energy cost $F_1$,

$$F_1 = E_a - T \ln g_1,$$

where $g_1$ is the number of available inequivalent $T_1$ processes. This energy cost $E_a$ is of the order of breaking through a local cage and the process involves only a few particles. Due to the amorphous nature of the supercooled liquid $E_a$ is a stochastic variable but the dispersion around $E_a$ is relatively small. We denoted the variance in the distribution of $E_a$ as $\sigma^2$ and noted that in this regime $\sigma/E_a \ll 1$.

(ii) The fact that $F_1$ does not change with the temperature in the Arrhenius regime indicates that $g_1$ is of the order of unity.

(iii) The precise reason for the blocking of this elementary relaxation channel when temperature is lowered is not known. We will propose, however, that this blocking is due to the disappearance of available free volume per particle $v_f(T, P)$ [12]. In other words, denoting by
the minimal average volume that every particle occupies, there must be additional free volume \( v_f \) for the \( T_1 \) process to occur.

Of course, the average volume per particle, which is the inverse of the density, \( v(T, P) \), is determined by the equation of state which varies from one material to the other. We will define \( T_f \) as the temperature where the free volume disappears, and then \( v(T_f, P) = v_m \). For the purposes of the present model we estimate the average volume per particle from the first order Taylor expansion:

\[
v(T, P) \approx v(T_f, P) + \left( \frac{\partial v}{\partial T} \right)_P (T - T_f) ,
\]

where the partial derivative is computed at \( T = T_f \). Accordingly, since \( v(T_f, P) = v_m \),

\[
v_f(T, P) \approx \left( \frac{\partial v}{\partial T} \right)_P (T - T_f) , \text{ provided } v > v_m \quad \text{(17)}
\]

\[
v_f(T, P) \approx 0 \text{ otherwise} . \quad \text{(18)}
\]

The probability of having a local free volume \( \tilde{v}_f \) is estimated from maximum entropy considerations as

\[
P(\tilde{v}_f) = \frac{1}{\tilde{v}_f} \exp \left( -\frac{\tilde{v}_f}{v_f} \right) . \quad \text{(19)}
\]

Consequently the probability that \( \tilde{v}_f > v_m \), which is the probability that a \( T_1 \) process could occur, is

\[
P(\tilde{v}_f > v_m) = \int_{v_m}^{\infty} P(\tilde{v}_f) d\tilde{v}_f = e^{-v_m/v_f} . \quad \text{(20)}
\]

Finally, we write \( v_f = v - v_m = \alpha v_m (T - T_f) \), where \( \alpha \) is the isobaric coefficient of thermal expansion at \( T = T_f \),

\[
\alpha = \frac{1}{v_m} \left( \frac{\partial v}{\partial T} \right)_P (T = T_f) . \quad \text{(21)}
\]

With this we can write

\[
P(\tilde{v}_f > v_m) = \exp \left[ -c \frac{T_f}{(T - T_f)} \right] , \quad \text{(22)}
\]

where the dimensionless constant \( c = 1/\alpha T_f \). Note that we can immediately estimate the probability of seeing the cooperative mechanism (which we will refer to as the \( T_2 \) process) as simply \( 1 - \exp \left[ -c \frac{T_f}{(T - T_f)} \right] \). Therefore we can estimate \( T^* \) where the two processes are equiprobable from

\[
\exp \left[ -c \frac{T_f}{(T^* - T_f)} \right] \approx 1/2 , \quad \text{(23)}
\]

and therefore \( T^* \approx (1 + c/\ln 2) T_f \).

(iv) In the fragile regime we propose that Eq. \( 15 \) retains its form, but now with a temperature dependent \( E_2(T) \) and a sizeable degeneracy \( g_2(T) \). Our picture is that of a relaxation process that involves more and more cooperative motions of a larger and larger number of particles. This obviously results in an increasing energy barrier \( E_2(T) \) which is indeed higher than \( E_\alpha \) but selected because the \( T_1 \) process is no longer available or it requires cooperation from the surroundings to be realized. Also an increase in the number of particles involved must increase \( g_2 \) simply due to the number of ways that these particles can be chosen. Thus we write in the fragile regime

\[
F_2(T) = E_2(T) - T \ln g_2(T) . \quad \text{(24)}
\]

We now consider the emergence of the \( T_2 \) process as a consequence of the disappearing of the \( T_1 \) process as is presented for example in Fig. \( 4 \) We write

\[
P(F, T) = \frac{W_1(F, T) + W_2(F, T)}{Z(T)} , \quad \text{(25)}
\]

\[
Z(T) = \int \left[ W_1(F, T) + W_2(F, T) \right] dF . \quad \text{(26)}
\]

Here \( W_1 \) and \( W_2 \) are the weights of the two relaxation channels respectively. In the Arrhenius regime \( W_1 \) is composed of three factors; remembering that \( g_1 \approx 1 \), we write in that regime

\[
W_1(F, T) = e^{-c \frac{T_f}{(T - T_f)}} \times \frac{1}{\sqrt{2 \pi \sigma^2(T)}} e^{-\frac{(F - \langle F \rangle)^2}{2 \sigma^2(T)}} \times e^{-F/T} . \quad \text{(27)}
\]

The first factor is the probability to have a \( T_1 \) process, Eq. \( 22 \) The second factor is the probability to select a free energy barrier of magnitude \( F \). The third factor is the probability to overcome that particular barrier. In the larger temperature regime that includes the pre- and post-Arrhenius regime we generalize Eq. \( 27 \) to read

\[
W_1(F, T) = e^{-c \frac{T_f}{(T - T_f)}} \times \frac{1}{\sqrt{2 \pi \sigma^2(T)}} e^{-\frac{(F - \langle F \rangle)^2(T)}{2 \sigma^2(T)}} \times e^{-F/T} . \quad \text{(28)}
\]
where $\sigma_1^2(T) \approx \sigma^2 + K(T - T_{Arr})^2$. This fit is supported by the scaling function shown in Fig. 9 in the left lower panel, and $T_{Arr}$ is the temperature of the minimum of scaling function.

Similarly,

$$W_2(F, T) = \frac{1 - e^{-cT_f/(T - T_f)}}{\sqrt{2\pi\sigma_1^2(T)}} e^{-\frac{(F - \langle F \rangle(T))^2}{2\sigma_1^2(T)}} \times e^{-F/T}$$

(29)

In Fig. 12 we present the prediction of Eq. (25) as a function of temperature. We chose the various parameters to agree with the functions shown in Fig. 9 with $c = 0.6$ and $T^* = 0.55$. The model indeed reproduces very well the main observations discussed above. We see the pre-Arrhenius behavior and its transition to Arrhenius behavior where expected. We also observe the rapid decline of the Arrhenius gaussian with the concurrent increase of the fragile gaussian which is marching to the right with increasing dispersion. We should stress that this simple model is not meant to represent the full complexity of the observed processes, especially at low temperatures where the cooperative dynamics reigns supreme; the point of this simple model is to stress the fundamental observation of this paper, i.e. that a compact and well defined relaxation mechanism appears to operate in the Arrhenius regime, whereas this simple mechanism is blocked once temperature reduces, giving rise to another mechanism which is presumably cooperative and much more involved. For all that we know one may encounter additional relaxation processes at still lower temperatures which are currently not available to numerical simulations.

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