Coexistence pressure for a martensitic transformation from theory and experiment: Revisiting the bcc-hcp transition of iron under pressure

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Coexistence pressure for a martensitic transformation from theory and experiment: revisiting the bcc-hcp transition of iron under pressure

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The coexistence pressure of two phases is a well-defined point at fixed temperature. In experiment, however, due to non-hydrostatic stresses and a stress-dependent potential energy barrier, different measurements yield different ranges of pressure with a hysteresis. Accounting for these effects, we propose an inequality for comparison of the theoretical value to a plurality of measured intervals. We revisit decades of pressure experiments on the bcc ↔ hcp transformations in iron, which are sensitive to non-hydrostatic conditions and sample size. From electronic-structure calculations, we find a bcc ↔ hcp coexistence pressure of 8.4 GPa. We construct the equation of state for competing phases under hydrostatic pressure, compare to experiments and other calculations, and address the observed pressure hysteresis and range of onset pressures of the nucleating phase.

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I. INTRODUCTION

Knowledge of the equation of state is crucially important in materials science and engineering, metallurgy, geophysics, and planetary sciences. However, equilibrium coexistence of phases during a pressure-induced martensitic transformation is extremely difficult to realize experimentally, and most shock and anvil cell experiments contain various amounts of a non-hydrostatic, anisotropic stress. Hence, an improved understanding of transformations arises when we can better compare idealized theoretical results to realistic experimental data. A long-studied case is iron (Fe), our focus below, but the results remain quite general.

Iron is the most stable element produced by nuclear reactions at ambient pressure, and one of the most abundant elements in the Earth. Thus, magneto-structural transformations1–28 and high-pressure states29–69 in iron attract enormous interest, especially in geophysics because iron is a primary constituent of the Earth’s core,29–163 many meteorites,163–172 and, due to its properties and availability, most steels.173 At low pressure P and temperature T, the α-phase of iron is a ferromagnet (FM) with the body-centered cubic (bcc) structure. At higher pressures, iron transforms to the ε-phase with hexagonal close-packed (hcp) structure of higher density that is non-magnetic or weakly anti-ferromagnetic. This transformation is martensitic,3 and the bcc-hcp equilibrium coexistence pressure is difficult to determine unambiguously experimentally (Table 1).

A martensitic transformation between bcc (α) and hcp (ε) phases can be characterized by four pressures (Table 1): a start and end pressure of direct α → ε (Pα→ε start, Pα→ε end) and reverse ε → α (Pε→α start, Pε→α end) transformations. Because martensitic stress is present in the anisotropic hcp phase but not in the isotropic bcc phase, we suggest the inequality

\[ P_{\text{end}}^{\epsilon \to \alpha} < P_0 < P_{\text{start}}^{\epsilon \to \alpha} \]  

for the \( \alpha - \epsilon \) equilibrium coexistence pressure \( P_0 \) and the observed hysteresis, rather than an inaccurate simple average3 \( P_{\text{avg}}^{\epsilon \to \alpha} = \frac{1}{2} (P_{\text{start}}^{\epsilon \to \alpha} + P_{\text{end}}^{\epsilon \to \alpha}) \).

The inequality (1) is applicable to all first-order structural transformations from an isotropic phase (e.g., Fe α-phase) with well-defined and reliably measurable \( P \) and \( T \) to one more prone to measurement error (e.g., anisotropic Fe ε-phase). This inequality is a generalization of what is expected in the classic liquid-gas isotherm in Fig. 1 from the van der Waals equation of state (EoS).174 If \( P \) and \( T \) are reliably measured in gas (G), but not in liquid (L) phase, then the isothermal equilibrium coexistence pressure \( P_0 \) at a given \( T \) is bounded by \( P_{\text{end}}^{L \to G} < P_0 < P_{\text{start}}^{G \to L} \), which is identical to (1). From the EoS,174 the overheated liquid and overcooled gas are observed during fast adia-

![FIG. 1: van der Waals isotherm (solid black), along with the unstable region (dashed yellow), and metastable overheated liquid and overcooled gas (blue). Delayed start of condensation and end of evaporation (dashed green arrows). Liquid and gas are indistinguishable at the critical isotherm (red) above the phase segregated region (dashed black line).](image-url)
batic processes or in a clean medium deprived of defects, able to serve as centers of evaporation and condensation. Due to heat of evaporation and condensation, and a limited thermal conductivity, it takes a very long time to maintain a constant temperature in both phases. In a faster process, start of condensation is delayed, and evaporation produces a cooled gas; hence, $P^L_H$ is on a lower isotherm, as in Fig. 1. Inequality (1) is applicable both to solid-solid and liquid-gas transformations, which have a hysteresis due to nucleation barriers.

For iron, while shock and anvil-cell (AC) pressure experiments give different averages $P_{\text{avg.}}$, they satisfy the more appropriate inequality (1), see Table I and Fig. 2d. To make direct and quantitative connections, we calculate the hydrostatic EoS of $\alpha$ and $\varepsilon$ Fe, determine $P_0$ via common-tangent construction, which should be thermodynamically relevant to purely hydrostatic equilibrium, and compare the result to experiment.

### II. BACKGROUND

**Previous Experiments:** Shock and AC pressure experiments are the major approaches to measure pressure-induced transformations, although hydrostatic conditions are often difficult to assess. Experimental onset (start) and final (end) pressures for $\alpha \rightarrow \varepsilon$ and $\varepsilon \rightarrow \alpha$ transformations are summarized in Table 1, which show a large span and the reason to revisit this issue.

For completeness, we highlight the experiments and their outcome for iron. Bancroft et al. studied propagation of compressive waves generated by high explosive in Artnco iron and reported a polymorphic transition at 13.1 GPa. Balchan and Drickamer used a high-pressure electrical resistance cell and found a sharp rise in resistance of iron at 13.3 GPa. Giles et al. showed that this bcc-hcp transformation is martensitic; their estimate of $P_0$ by $P_{\text{avg.}} = 10.7 \pm 0.8$ GPa differs from the earlier reported $P_{\text{start}} = 13$ GPa, often quoted as the martensitic start pressure. Mao, Bassett, and Takahashi performed XRD measurements of lattice parameters of iron at 23°C at pressures up to 30 GPa, and suggested a bcc-hcp shear-shuffle model (their Fig. 3 is reproduced in Ref. 6). Bassett and Huang applied a non-hydrostatic pressure with an uncontrolled shear strain (known to produce pressure self-multiplication) and confirmed an atomic mechanism of the bcc-hcp transition, but omitted discussion of changes in volume and magnetization in their shear-shuffle model. Zou et al. used solid He as the pressure medium in their diamond AC (DAC) experiments on iron (99.95 wt. % Fe) powder pressed into a plate and on a folded section of a 10 micron foil; they pointed at the uniform non-hydrostatic stress as a possible cause of differing data.

Importantly, transition pressure estimates depend on how hydrostatic the applied stress is and sample size. For example, Bargen and Boehler found that the pressure interval of the forward bcc→hcp transition increases with increasing non-hydrostaticity (transition pressures and hysteresis width change systematically with the shear strength of the pressure medium). The best pressure medium is a superfluid; a good one is a gas or a fluid with a low viscosity; the worst one is a viscous fluid or a solid. Due to grain boundaries and melting-freezing waves, solid helium (He) can behave as a superfluid.

Taylor et al. focused on the large hysteresis and used a DAC up to 24 GPa; as pressure is increased, Fe is fully converted to hcp at $P_{\text{start}} \approx 23$ GPa. Upon reducing pressure, half of the hcp transforms to bcc by $P_{\text{avg.}} = 9.5$ GPa, while a small $\varepsilon$-Fe remnant is present at $P_{\text{avg.}} = 7.7$ GPa. They report $P_{\text{avg.}}$ values from 8.6 to 15 GPa. Using a radial diffraction DAC with infrared laser heating on Alfa Aesar (99.9% pure) Fe powder ($10^{-5}$ m particle size), Miyagi et al. reported appearance of hcp at 10 GPa that fully converts near 22 GPa, while bcc appears at 8 GPa during decompression. Jiang et al. studied grain-size and alloying effects on the transition pressure, finding that $P_{\text{start}}$ shifts from 13 GPa in bulk to 11 GPa in nano-crystalline samples. Wang, Ingalls, and Crozier performed an XAFS study at 23°C up to 21.5 GPa; a mixed-phase region was found between $P_{\text{start}} = 13$ and $P_{\text{end}} = 20$ GPa, and between $P_{\text{start}} = 15$ and $P_{\text{end}} = 11$ GPa. Later, Wang and Ingalls used XAFS with a sintered boron-carbide anvil cell to measure lattice constants and bcc abundance versus $P$, and reported $P_{\text{avg.}} = 13$ GPa and $6.6 \leq P_{\text{end}} \leq 8.9$ GPa. Using in situ EXAFS measurements and nanosecond laser shocks, Yaakobi et al. detected hcp phase and claimed that the $\alpha \rightarrow \varepsilon$ transition can happen very quickly.

Finally, the change of magnetization along a transition path is important, where there is an abrupt 8–10% volume decrease at the transition state. Former bcc-hcp equilibrium pressure calculations provided values of 13.1 GPa, 11 GPa, 10.5 GPa, and 10 GPa in apparent agreement with the experimental values of $P_{\text{start}} = 13$ GPa and $P_{\text{avg.}} = 10.7 \pm 0.8$ GPa. However, those calculated pressures disagree with later experimental data (Table 1). Using ab initio molecular dynamics (MD), Belonoshko et al. considered shear at the Earth’s core conditions and constructed an EoS for $\alpha$ and $\varepsilon$ Fe. Wang et al. studied nucleation of the higher-pressure hcp and fcc phases by classical MD simulations employing an embedded atom method (EAM) potential, and found that the transformation happens on a picosecond timescale; their calculated transition pressure is around 31–33 GPa for uniform and 14 GPa for...
TABLE I: Start and end pressures [GPa] with width $\Delta P$ = $|P_{\text{end}} - P_{\text{start}}|$ for iron bcc ($\alpha$) $\rightarrow$ hcp ($\varepsilon$) direct and inverse transformations. Type of experiment (Expt.) specifies shock or anvil cell (AC), form of sample (bulk, foil, powder), or pressure medium (He, Ar, m-e for methanol-ethanol, etc.) in the AC. The $P_{1/2}^{\alpha\varepsilon}$ values at half-transition (50% bcc + 50% hcp) are in the square brackets [$P_{\text{start}}$ column].

| Ref. | Year | $P_{\text{start}}^{\alpha}$ | $P_{\text{end}}^{\varepsilon}$ | $\Delta P^{\varepsilon\rightarrow\alpha}$ | $P_{\text{start}}^{\varepsilon}$ | $P_{\text{end}}^{\alpha}$ | $\Delta P^{\alpha\rightarrow\varepsilon}$ | Expt. |
|------|------|-----------------|-----------------|-------------------------------|-----------------|-----------------|-------------------------------|-------|
| 1    | 1956 | 13.1            |                 |                               |                 |                 |                               | shock |
| 2    | 1961 | 13.3            |                 |                               |                 |                 |                               | resistance |
| 3    | 1971 | 13.3            | 16.3            | 3                             | 8.1             | 4.5             | 3.6                           | AC    |
| 4    | 1981 | 13.52           | 15.27           | $<2$                          | 9.2             | 6.74            | 2.5                           | powder |
| 5    |      | 15.21           | 15.47           | $<1$                          | 10.23           | 8.5(6)          | 2                             | foil  |
| 6    | 1987 | 10.8            | 21              | $\approx10$                   | 15.8            | 3               | 13                            | Au    |
| 7    | 1990 | 10.6            | 25.4            | 14.8                          | 16              | 4               | 12                            | Al2O3  |
| 8    |      | 10.7            | 21.6            | 10.9                          | 16.2            | 3.7             | 12.5                          | Au    |
| 9    |      | 12.4            | 17.8            | 5.4                           | 12.2            | 4.8             | 7.4                           | NaCl  |
| 10   |      | 12.8            | 17.2            | 4.4                           | 11.8            | 5.5             | 6.3                           | CsI   |
| 11   |      | 14.3            | 17.5            | 3.2                           | 11.9            | 7               | 5                             | m-e   |
| 12   |      | 14.9            | $<15.9$         | 0.5                           | $<11$           | $>7$             | $<4$                          | Ar    |
| 13   |      | 15.3            | 15.3            | 0.1                           | 10.6            | 8.0(6)          | 2                             | He    |
| 14   |      | 8.6             | 23              | $\approx14$                   | [9.5]           | 7.7             | 3.6                           | hydrostatic |
| 15   | 1998 | 13.0            | 18.6            | $\approx5.6$                  | [10.3]          | 6.6             | 7.4                           | XAFS  |
| 16   | 2001 | 13              | 17              | $\approx4$                    | 8               | 5               | 3                             | bulk  |
| 17   |      | 11              | 14              | $\approx3$                    | 7               | 1               | 6                             | nano-Fe |
| 18   | 2005 | 14              | 16              | 2.4                           |                 |                 |                               | AC    |
| 19   | 2008 | 10              | 22              | $\approx12$                   | 8               | 4               | 4                             | powder |

uniaxial compression (but there is no magnetization in the EAM potential). Caspersen et al.\textsuperscript{26} showed that presence of a modest shear accounts for the scatter in measured transformation pressures, affecting the hysteresis. Johnson and Carter\textsuperscript{19} used a drag method in a rapid-nuclear-motion (RNM) approximation and obtained an unphysical discontinuous jump in atomic shuffle degrees of freedom, giving a very low bcc-hcp barrier; they found that bcc and hcp phases have equal enthalpies at the calculated pressure of 13.1 GPa.

Liu and Johnson\textsuperscript{20} directly constructed the potential energy surface in a 2-atom cell for the shear-shuffle model,\textsuperscript{6} allowing changes of lattice constants and (continuous) atomic degrees of freedom; although hydrostatic pressure cannot produce shear, pressure does affect the potential energy surface and barriers. They reported $\approx9$ GPa for bcc-hcp coexistence; the calculated kinetic barriers along the transition path were 132 meV/atom at 0 GPa with an estimated minimum (maximum) onset pressure of 9 (12.6) GPa, 119 meV/atom at 10.5 GPa with a min (max) onset at 8.1 (13.8) GPa, and 96 meV/atom at 22 GPa with a min (max) onset of 6.6 (10.2) GPa. That is, there is an expected 3.6 to 5.7 GPa hysteresis width depending on kinetic pathway (and volume fluctuations). In addition, they showed that drag methods decouple degrees of freedom incorrectly, as confirmed later.\textsuperscript{184}

Recently, Dupe et al.\textsuperscript{186} reconsidered the transition mechanism within the same shear-shuffle model,\textsuperscript{6} but incorrectly fixed the volume at 71.5 bohr$^3$/atom (no volume collapse allowed), and used the RNM drag method to compare energies of three shuffling mechanisms at constant shear and volume. Friak and Sol\textsuperscript{24} in a 4-atom cell considered non-magnetic (NM) and antiferromagnetic (AFM) orderings along a predefined path (which were almost degenerate); their energy-volume common-tangent gave coexistence $P_0$ at 10.5 GPa.\textsuperscript{24}

III. PRESENT RESULTS

To determine $P_0$ of equilibrium coexistence of FM bcc and NM hcp phases, we calculate volume $V$, energy $E$, and enthalpy $H = E + PV$ (Fig. 2) at various hydrostatic external pressures $P$. Each unit cell is fully relaxed at a given $P$. All atomic forces and all non-diagonal pressure components remain zero due to symmetry. Diagonal pressure components are the same by symmetry in bcc and fcc phases, while their difference does not exceed 0.03 GPa in hcp. Magnetization of the FM bcc reduces with pressure and collapses to zero at $\approx900$ GPa; hcp magnetization is set to zero at all pressures.

The slope of the common tangent to the $E(V)$ curves in Fig. 2a gives $P_0$ of 8.4 GPa; this pressure gives zero enthalpy difference in Fig. 2c, and is compared to all experiments in Fig. 2d. The previously calculated values of 13.1 GPa\textsuperscript{19} and 10.5 GPa\textsuperscript{24} do not agree with all the experimental data, summarized in Table I and Fig. 2d.

To obtain these results, we used the Vienna ab initio simulation package (VASP)\textsuperscript{187–189} with generalized gradient approximation (GGA)\textsuperscript{190,191} and projector augmented-wave (PAW) potentials.\textsuperscript{192,193} We use 334.88 eV energy cutoff for the plane-wave basis with augmentation charge cutoff of 511.4 eV. The modified Broyden method\textsuperscript{194} is used for self-consistency. We carefully check
There are many EoS for solids. The role of the exchange correlation functional $V_{xc}$ is well discussed in the literature. Previous failure of the LDA to properly describe the Fe ground state was an impetus in developing the GGA, which we use. We rely on decades of DFT testing, and appearance of the impetus in developing the GGA, the LDA to properly describe the Fe ground state was an well discussed in the literature.

The parameters are given in Table 2 for FM bcc at low pressure, and NM hcp at high pressure. Although hcp at lower pressure and density ($V > 23 \text{ Å}^3$/cell) changes from NM to AFM, their $E(V)$ curves at $V < 21 \text{ Å}^3$ are almost degenerate. These values have some dependence on the range of fitted data, and are affected by the EoS functional form. As expected, calculated volume $V_0$ is reduced by 3% compared to experiment due to the standard DFT systematic error (i.e., 1% in lattice constants). This DFT error introduces a systematic 3% error (0.25 GPa) in our bcc-hcp coexistence pressure.

Our result for bcc iron is in agreement with the previous DFT calculations, with $B_0$ ranging from 171 to 194 GPa from EMTO, VASP, and Wien2K codes, which compare well with the assesses values of 159–206 GPa. Due to their pressure and temperature dependence, the assessed elastic constants and EoS parameters depend on the fitting procedure. The measured by inelastic neutron scattering at 300 K bulk modulus $B = C_{11} - \frac{2}{3}C'$ changes from 159 GPa at zero pressure to 206 GPa at $P = 9.8$ GPa. Our EoS coefficients for the hcp single crystal agree with previously calculated

FIG. 2: (a) Energy [meV/atom] relative to bcc at 0 GPa and (b) pressure [GPa] versus volume [Å³/cell] for hydrostatically relaxed 2-atom unit cells of bcc FM (black), hcp NM (blue) and AFM (green), and fcc NM iron (orange); with DFT values (dots) and least-squares fit to the Birch-Murnaghan EoS (lines). Common tangent construction (red line) yields $P_0 = 8.42$ GPa. Vertical dotted lines are guides to the eye. (c) Enthalpy difference [meV/Fe] between hcp and bcc phases versus pressure [GPa] (dots) and least-squares fit to the Birch-Murnaghan EoS (lines). Common tangent construction (red line) yields $P_0 = 8.42$ GPa. Vertical dotted lines are guides to the eye. (d) Comparison of the calculated $P_0$ = 8.42 GPa (vertical red line) with experimental data from Table I, represented by $[P_{end} - P_{start}]$ horizontal segments. Except for the 1991 hydrostatic experiment, most diamond anvil cells provided uniaxial or highly anisotropic pressure.

convergence with respect to the number of $k$-points (up to $32^3$=32768) in the Γ-centered Monkhorst-Pack mesh within the tetrahedron method with Blöchl corrections. Gaussian smearing with $\sigma = 0.05$ eV with $16^3$=4096 $k$-points in the 2-atom cell is used for relaxation.

There are many EoS for solids. We fit our $E(V)$ data in Fig. 2a to the Birch-Murnaghan EoS. For iron, the parameters are given in Table 2 for FM bcc at low pressure, and NM hcp at high pressure. Although hcp at lower pressure and density ($V > 23 \text{ Å}^3$/cell) changes from NM to AFM, their $E(V)$ curves at $V < 21 \text{ Å}^3$ are almost degenerate. These values have some dependence on the range of fitted data, and are affected by the EoS functional form. As expected, calculated volume $V_0$ is reduced by 3% compared to experiment due to the standard DFT systematic error (i.e., 1% in lattice constants). This DFT error introduces a systematic 3% error (0.25 GPa) in our bcc-hcp coexistence pressure.
TABLE II: Birch-Murnaghan EoS parameters, assessed or extrapolated from theory and experiment. Values depend on data, its $P$ and $T$ range, and fitting procedure, see text.

| Fe phase | $V_0$ $\text{Å}^3$ | $B_0$ GPa | $B'_0$ |
|----------|-----------------|-----------|--------|
| bcc FM   | 22.72 6.84      | 185       | 4.7    |
| bcc expt. | 23.56 7.09      | 172       | 5.0    |
| bcc expts. | 159-206        |           |        |
| hcp NM   | 20.34 6.13      | 293       | 4.5    |
| hcp AFM  | 19.94 6.004     | 140       | 3.9    |
| hcp expt.$^a$ | 22.35 6.73    | 167       | 5.8    |
| hcp expts. | 156-195-4       | 119-150   | 4-6    |

ones$^{204,205}$ at $T=0$ K, summarized in Table 1 in Ref. 204. However, the experimentally assessed EoS for the hcp martensite$^{82,206}$ with $B_0$ of 156–195 GPa and $B'_0$ of 4.3–5.8 differs from that calculated for a hcp single crystal (Table II). This difference is expected because a martensite is a composite with both compressed and dilated regions, and any non-homogeneous distortion increases energy, shifting up and distorting the $E(V)$ curve in Fig. 2a. We note that although the fitted values of $B_0$ (293 and 140 GPa in Table 2) differ quite dramatically for the hcp NM and AFM curves due to their divergence at larger $V$ and lower pressures, these curves coincide near $P_0$ and their tangent is the same in Fig. 2a.

In addition, we use the generalized solid-state nudged elastic band (GSS-NEB) method$^{184,185}$ to find the transformation path at $P_0$ within the shear-shuffle model,$^6$ see Fig. 3. Enthalpy of the transition state (TS) is 156 meV/Fe above the bcc and hcp, which have the same enthalpy at $P_0$. Coherent coexistence of the magnetic and non-magnetic states with the same atomic and cell degrees of freedom at the TS, which has a form of a cusp, causes a coherency stress, related to the 25 GPa discontinuity in the internal pressure in Fig. 3. The electronic interactions propagate with the speed of light, while pressure propagates with the speed of sound. Hence, electronic transitions (causing discontinuity in magnetization in Fig. 3) happen faster than the structural transformation (resulting in continuous volume, shuffle and shear change within the GSS-NEB method) during the magneto-structural transformation of iron under pressure. To state it briefly, change of density (or unit cell volume $V$) causes transformation of the magnetic state and a discontinuous change of magnetization $M$, which creates a discontinuous change of the internal pressure $P$, which pushes further adjustment of the volume; the causality is $V \rightarrow M \rightarrow P \rightarrow V$.

Due to an energy barrier (Fig. 3), there is a hysteresis, and the forward $\alpha \rightarrow \varepsilon$ transformation starts at a higher pressure than $P_0$. The austenitic bcc $\alpha$ phase has a chance of being isotropic at hydrostatic pressure. However, the anisotropic hcp $\varepsilon$-phase forms a martensite, which has non-hydrostatic internal stress in experiment; due to the difference between hexagonal $a$ and $c$ directions it is not isotropic at the microscale. Only when this phase disappears, one can make a comparison between theoretical and experimental hydrostatic pressure.

A question remains: “Why should a common-tangent construction for the ideal relaxed structure be relevant to experiments at all?” In Fig. 2a we deal with energies of the ideal relaxed structures at hydrostatic pressure, while there are imperfections in experiments. At any hydrostatic pressure $P$, energy of the relaxed ideal structure is the lowest; defects and non-hydrostaticity effects shift it up, and we can predict the consequence of this shift for direct $\alpha \rightarrow \varepsilon$ transformation. Martensite has more structural defects and anisotropy compared to austenite, giving a larger systematic upward shift for the hcp phase of the energy curves in Fig. 2a and a larger slope (higher pressure) for the common-tangent. In addition to an energy barrier, this effect partially explains a tendency towards higher $P_{\alpha \rightarrow \varepsilon }^{\text{end}}$ values in less hydrostatic experiments (Fig. 2d). Again, a martensite has anisotropic martensitic pressure (up to 25 GPa from Fig. 3), hence one can deal with a hydrostatic pressure only after complete disappearance of the martensitic phase, which is why $P_{\alpha \rightarrow \varepsilon }^{\text{end}}$ brackets $P_0$ in (1) on the lower side. The enthalpy barrier (Fig. 3) is responsible for the fact of the hysteresis, and a variety of structural defects and non-hydrostatic contributions in experiments accounts for the scatter of experimental data (Fig. 2d). We emphasize the import of (1) for comparing an ideal value of $P_0$ to all experiments.

FIG. 3: Enthalpy $H$, volume $V$, magnetization $M$ [per atom], internal pressure $P$, atomic shuffle [defined as direct coordinate of the second atom in a 2-atom unit cell relative to the first atom placed at zero], and shear angle [degrees] at $P_0$ versus the GSS-NEB path from bcc to hcp.
IV. DISCUSSION

Iron: Transformation from \( \alpha \) (bcc) to \( \varepsilon \) (hcp) iron is martensitic,\(^5\) and the hysteresis loop can be characterized by four pressures: \( P^{\alpha \rightarrow \varepsilon}_{\text{start}} \), \( P^{\alpha \rightarrow \varepsilon}_{\text{end}} \), \( P^{\varepsilon \rightarrow \alpha}_{\text{start}} \), and \( P^{\varepsilon \rightarrow \alpha}_{\text{end}} \). In experiment,\(^5,7\) \( \varepsilon \)-phase appears at \( P^{\varepsilon \rightarrow \alpha}_{\text{start}} \) between 8.6 and 15.3 GPa, while \( \alpha \)-phase is fully converted above \( P^{\alpha \rightarrow \varepsilon}_{\text{end}} \) between 14 and 25 GPa upon loading. Whereas, upon unloading, \( \alpha \)-phase appears at \( P^{\alpha \rightarrow \varepsilon}_{\text{start}} \) between 16 and 7 GPa and \( \varepsilon \)-phase disappears below \( P^{\varepsilon \rightarrow \alpha}_{\text{end}} \) between 8 and 1 GPa. Importantly, there is no strict inequality between \( P^{\alpha \rightarrow \varepsilon}_{\text{start}} \) and \( P^{\varepsilon \rightarrow \alpha}_{\text{start}} \) due to the martensitic stress distribution in the \( \varepsilon \)-phase.

Our calculated \( P_0 \) of 8.4 GPa is below \( P^{\alpha \rightarrow \varepsilon}_{\text{start}} \) and above \( P^{\varepsilon \rightarrow \alpha}_{\text{end}} \), see inequality (1). It agrees well with the experimental distribution of \( P^{\alpha \rightarrow \varepsilon}_{\text{start}} \geq 8.6 \) GPa and \( P^{\varepsilon \rightarrow \alpha}_{\text{end}} \leq 8.5 \) GPa. The observed \( P^{\alpha \rightarrow \varepsilon}_{\text{start}} \) and \( P^{\varepsilon \rightarrow \alpha}_{\text{end}} \) are highly affected by the martensitic stress within the hcp \( \varepsilon \)-phase. A martensitic transformation occurs between an isotropic (bcc) austenite and an anisotropic (hcp) martensite, which experience martensitic stress resulting in anisotropic distortions. In other words, there is little internal stress in austenite and large anisotropic internal stresses in martensite. However, martensitic stress is not taken into account in our calculation of the bcc-hcp equilibrium coexistence pressure \( P_0 \). Because hcp does not exist below \( P^{\alpha \rightarrow \varepsilon}_{\text{start}} \) and \( P^{\varepsilon \rightarrow \alpha}_{\text{end}} \), these values should not be affected by the martensitic stress in hcp (though the transformation can be delayed due to an energy barrier), and can be used in the proper comparison to experiment. Hence, \( P_0 \) must be between \( P^{\alpha \rightarrow \varepsilon}_{\text{start}} \) and \( P^{\varepsilon \rightarrow \alpha}_{\text{end}} \), see inequality (1). These experimental ranges are compared with our calculated value of \( P_0 \) in Fig. 2d, with an excellent agreement between theory and experiment.

In Fig. 3 we present the energy barrier and the transition state at \( P_0 \) via a generalized solid-solid nudged-elastic band\(^{184}\) method that incorporates magnetization and volume collapse. Change of magnetization of the TS from FM to NM results in the pressure change by \( \Delta P = 25 \) GPa. This calculated \( \Delta P \) at the TS agrees with the observed bcc-hcp coexistence interval \([P^{\varepsilon \rightarrow \alpha}_{\text{end}}, P^{\alpha \rightarrow \varepsilon}_{\text{start}}]\).

Lithium: To illustrate further the generality of inequality (1), we apply it to the pressure-induced transformation in fcc (isotropic austenite) lithium (99.9\% \( 7\)Li) to the intermediate hR1 (rhombohedral) phase near 39 GPa.\(^{209,210}\) The hR1 structure with 1 atom/cell is a 7% distortion of fcc along a cubic axis; it can form a martensite due to axial anisotropy. At 180 K and increasing \( P \) in a DAC under inert gas,\(^{210}\) the structure remains fcc up to 38.3 GPa; a new set of XRD intense diffraction lines appears at \( P^{\text{bcc-hR1}} = 39.8 \) GPa (and disappears at \( P^{\text{hR1} \rightarrow \text{cI16}} = 42.5 \) GPa), thus providing an upper bound for \( P_0 < P^{\text{bcc-hR1}} \). Higher values for the upper bound are provided by dynamic shock compression experiments,\(^{209}\) which measure a sharp increase of resistivity at \( P^{\text{bcc-other}} = 42 \) GPa (isentropic compression from 77K) or 43 GPa (multistep shock at \( T < 660 \) K); the other phase was not identified. A comparable upper bound is offered by the DAC experiment\(^{211}\) at 25 K, which found a small resistivity drop at \( P^{\text{bcc-other}} = 42 \) GPa upon loading (attributed to transitions from fcc to hR1 and from hR1 to cI16). Interestingly, no change is observed in the Raman spectra at these pressures.\(^{212}\) While compression data provides \( P^{\text{bcc-hR1}} \) as an upper bound of \( P_0 \) for the fcc→hR1 transformation, which might be martensitic. Due to the hysteresis, current structure depends not only on the instant \( P \) and \( T \) values, but also on the path (history), and should be represented by a directed arrow (not just a dot) on the \( P-T \) diagram (e.g., Fig. 1 in Ref. 213).

V. SUMMARY

We presented and explored a methodology for comparing idealistic theoretical predictions at hydrostatic pressure to realistic experiments with anisotropic stress, based on inequality 1. Due to its importance, we revisited the iron bcc-hcp equilibrium coexistence; our calculated pressure of \( 8.4 \pm 0.5 \) GPa is in agreement with available experimental data. Anisotropic internal stress in the hcp martensite, difference in volume between FM bcc and NM (and competing AFM) hcp iron near the transition state contribute to the spread of the experimentally assessed (non-equilibrium) bcc-hcp coexistence pressures, as well as to the uncertainty in the equation of state of hcp martensite. We emphasized the difference between a single crystal and a martensite and improved understanding of the available data for iron under pressure. Importantly, we suggested a universal inequality (1), graphically illustrated in Fig. 2d, for proper comparison of the assessed and calculated pressures characterizing first-order transitions in many materials, in particular the martensitic transformations in solids, as shown for iron and lithium.

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