Study on the Effect of Plasma Treatment on Flat-band-voltage and Equivalent Oxide Thickness using Metal-organic Chemical Vapor Deposition TiN Film as p-MOSFETS Metal Gate Electrode

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Abstract

This work aimed to study on the effects of plasma treatment on flat band voltage (Vfb) and equivalent oxide thickness (EOT) using Metal-organic Chemical Vapor Deposition (MOCVD) TiN film as p-MOSFETS metal gate electrode. The plasma treatment conditions effect on the resistance and composition properties of MOCVD TiN, consequently, they can modulate work function and control threshold voltage. The effects of “plasma treatment” were imposed to favor the formation of Ti-N bonds, thus decreasing the rate Ti-C bond and favoring a crystallized stoichiometric TiN phase to increase PMOS Vfb shift. Meanwhile, plasma treatments accelerate interfacial oxide formation and increase EOT. On the other hand, EOT increases with the increase of plasma treatment time and power in this work, but Vfb does not always increase and reach a max value.

Keywords: MOCVD TiN; Vfb; Plasma treatment; pMOSFETs

Abbreviations: Vfb: Flat-Band-Voltage; EOT: Equivalent Oxide Thickness; MOS: Metal-Oxide-Semiconductor; TDMAT: Tetra (Dimethylamino) Titanium; XPS: X-Ray Photoelectron Spectrometer; AFM: Atomic Force Microscope; XRD: X-Ray Diffraction; TEMAH: Tetras (Ethylmethylamido) Hafnium; IL: Interfacial Layer; ALD: Atom Layer Deposition; HRTEM: High Resolution Transmission Electron Microscopy; FWHM: The Full Width Half Maximum (FWHM); RMS: Root Mean Square

Introduction

Scaling of sub-45 nm complementary metal-oxide semiconductor CMOS devices requires metal gate electrodes to address restricting factors of usual poly silicon gates, such as poly depletion, doping species penetration, and Fermi level pinning [1,2]. However, integration of the high K dielectric/metal gate stack into a metal-oxide-semiconductor (MOS) transistor provides some new challenges, such as Vfb and EOT control [3,4]. The requirements for the metal gates are as follows: favorable work functions, low sheet resistance, thermal stability, and compatibility with high-k dielectric Among all the potential candidates for metal gate electrodes, TiN gates showed promising results such as thermal stability with the high-k gate dielectrics and process compatibility [5,6]. Currently, Titanium nitride (TiN) films deposited with atom layer deposition (ALD) and reactive sputter are widely studied as metal gate in semi-conductor technology [5-9].

In this work, TiN films were deposited by using a MOCVD technique. The Tetra (dimethylamino) titanium (TDMAT) precursor has been used to obtain TiN films. N2 and H2 plasma treatment after TiN thin film deposition TiN film as p-MOSFETS metal gate electrode. The flow rates of H2, and N2 gases were 450 sccm and 300 sccm, respectively. The process pressures of the TiN deposition and the plasma treatment were at 5 torr and 1.3 torr, respectively. The sheet resistance of blanket TiN films was analyzed by a four-point probe. The film properties of MOCVD TiN films were examined by X-Ray Photoelectron Spectrometer (XPS), Atomic Force Microscope (AFM) and X-Ray Diffraction (XRD).

A MOS capacitor structure of metal SiO2/p-type Si substrate was fabricated as shown in Figure 1a. Interfacial layer (IL) was grown to ~0.8 nm with ozone chemical. 2 nm HfO2 film was deposited on gate oxide using Tetrakis (ethylmethylamido) hafnium (TEMAH) as the precursor and H2O as the oxidant at 300°C in TFS200, then, these sample underwent post dielectric annealing under 450°C, 15 s at N2 ambience with a trace of O2 at atmospheric pressure. TiN with 2 nm as...
capping layer was deposited on high-k layer. Next MOCVD TiN metal gates with different process conditions, which will be detailed in the following section, were deposited with Endura 5500 as work function layers. 750 Å ALD W cap layers were then deposited for all the samples. Finally, after back oxide removal, Al metallization and the formation of gas annealing are performed. Moreover, another MOS capacitor is fabricated as shown in Figure 1b, Ti O-scavenging layer was deposited between ALD TiN cap layer and MOCVD TiN film. Vfb and EOT for the capacitors were extracted from high frequency capacitance-voltage (C-V) measurements with a Keithley 4200-SCS analyzer at 1 MHz. High Resolution Transmission Electron Microscopy (HRTEM) was carried out to characterize the physical properties.

**Results and Discussion**

Figure 2 shows the impacts of plasma treatment on MOCVD TiN film resistance. As shown in Figure 2, there is very obvious impact on film resistance with plasma treatment, film resistance will obvious decrease with increasing plasma treatment time and power. At the same condition of plasma treatment time, the resistance of higher plasma power treatment is lower than the lower condition; when plasma treatment time reaches or exceeds 35 s, there is very little difference among 250 W, 400 W and 750 W plasma treatment conditions. XRD analysis results are shown in Figure 3 with different plasma treatment time at plasma treatment power of 400 W condition. It can be seen that the diffraction intensity is very weak without plasma treatment condition, and extending plasma treatment time makes MOCVD TiN film (200) preferential orientation more remarkable and the full width half maximum (FWHM) of (002) diffraction more narrow, it is known that the FWHM obtained from the result of XRD relates to finite crystallite sizes, stacking faults, micro twins, stresses, dislocations, concentration variations, which parameters have a strong dependence on the crystal quality of films [10]. This suggests that plasma treatment make atmosphere TiN film including high C and O concentration crystallize and form strong Ti-N bonds. The chemical composition of MOCVD TiN film was analyzed by X-Ray Photoelectron Spectrometer (XPS), the Ti2p peak can be fitted by eight Gaussian functions as Figure 4 shown, the XPS spectrum of Ti2p shows obviously different with and without plasma treatment. Compared with untreated MOCVD TiN films, Ti2p(TiN) ratio obviously increases and Ti2p(TiC, TiOx and TiO2)ratio decrease with plasma treatment, it is helpful to form Ti-N bonds using plasma treatment, plasma treatment decreases the Ti-C and Ti-O bond and favors a crystallized stoechiometric TiN phase. This result is matched with XRD analysis result. Figure 5 shows RMS roughness of MOCVD TiN with and without plasma treatment. It can be seen that plasma treatment can improve obviously MOCVD TiN film roughness which can be related with crystallization of TiN film due to decrease of C and O level observed on Figure 4a XPS result.

In order to investigate the effect of MOCVD TiN thickness on its work function, MOCVD TiN films with four different growth time (0 s, 4.1 s, 7 s, 15 s) were deposited, growth time 0 s condition do not treat with plasma, other three conditions treated with plasma(power 750 W and time 35 s). MOS capacitors are fabricated, and the C-V characteristics were measured for Vfb and EOT extraction. Figure 6 shows the impacts of MOCVD TiN film growth time on Vfb and EOT. Vfb shifts positively with the increase of MOCVD TiN film growth time, the Vfb of growth 7 s MOCVD TiN film is closely with growth 15 s MOCVD TiN film. This means that there is a critical thickness for the work function of MOCVD TiN film, it will stabilize over this critical
value. It is reported that amorphous and laminated film has a lower work function than its crystalline and bulk counterpart due to densely packed crystallographic surfaces and therefore fewer broken surface bonds [11-14]. As the thickness of TiN increase, the crystallinity increases until it reaches a critical thickness, then the work function of TiN will be constant due to stabilizing crystallinity [15]. It can also be seen from Figure 6 that the EOT decreases obviously with increasing growth time of MOCVD TiN film. The effects of plasma treatment were imposed to investigate the effect of plasma treatment on MOCVD TiN film work function and composition properties of the IL layer. The plasma treatment conditions effect on the resistance and concentration gradient of high energy oxygen atoms; this will decrease oxygen atoms diffusion length and decrease concentration gradient of high energy oxygen atom; this will decrease oxygen atoms diffusion flux and decreases EOT of MOS capacitors (Figure 7). Because oxygen atom cannot gain energy without plasma treatment and their diffusion will be suppressed (Figure 9c), it is very easy to explain why EOT of plasma untreated sample (c) is almost equal with initial IL thickness. Compared to the sample (a), there is thinner IL layer for sample (d) with Ti layer deposited between ALD TiN cap layer and MOCVD TiN film (Figure 8d), it was attributed to thin Ti metal film used as oxygen scavenging layer. In other words, Ti metal film acts as oxygen diffusion barrier and suppresses oxygen diffusion into IL layer (Figure 9d). From another point of view, it is proved that the increase of EOT with the increase of plasma treatment time and power is due to oxygen diffusion of MOCVD TiN film into IL layer.

**Summary and Conclusions**

Titanium nitride (TiN) films were prepared by using a metal-organic chemical vapor deposition (MOCVD) technique with varying growth and plasma treatment conditions. The plasma treatment conditions effect on the resistance and composition properties of MOCVD TiN film. The effects of plasma treatment were imposed to favor the formation of Ti-N bonds, thus decreasing the rate Ti-C bond and favoring a crystallized stoichiometric TiN phase to increase Vfb value as PMOS metal gate electrode. Meanwhile, plasma treatments with increasing plasma treatment time, and the impact of high power on EOT is more obvious than low power. Next, the impacts of plasma treatment on EOT will be discussed in detail.

Figure 8 shows the HRTEM images of samples as listed in Table 1. It can be seen that the SiO$_2$-like interfacial layer (IL) of sample (a) is the thickest in all samples, the IL thickness of samples (c and d) is closely with initial 0.8 nm thickness, and the IL thickness of sample (b) is thicker than sample (c and d), but thinner than sample (a). The model of the effect of plasma treatment on EOT is schematically illustrated in Figure 9. It is well known that elements diffusion ability is related with diffusion coefficient D and concentration gradient according to Fick’s first law of diffusion, and D is related with temperature. There is high concentration oxygen atom in plasma untreated MOCVD TiN film as shown in Figure 4b, the oxygen atoms of the film surface that gain energy from plasma have strong diffusion ability, and they are easily diffuse to HfO$_2$/interface layer when plasma treatment time is due to oxygen diffusion of MOCVD TiN film into IL layer. These oxygen vacancies in the HfO$_2$ layer act as mediators for oxygen atoms diffuse across the HfO$_2$/IL layer. These oxygen atoms react with substrate Si atom and form SiO$_2$ that increase EOT of MOS capacitors (Figure 9a). It is reported that oxygen diffusion through amorphous HfO$_2$ is quite rapid and this is not a rate-limiting step for this process [14]. Increasing MOCVD TiN thickness will increase oxygen atoms diffusion length and decrease concentration gradient of high energy oxygen atom; this will decrease oxygen atoms diffusion flux and decreases EOT of MOS capacitors (Figure 9b). This is consistent with our results in Figure 6, which shows the EOT decreases obviously with increasing growth time of MOCVD TiN film. In addition, increasing MOCVD TiN plasma treatment time will increase oxygen atom diffusion time, which makes more oxygen atoms diffuse into IL, this will increase EOT of MOS capacitors (Figure 7). Because oxygen atom cannot gain energy without plasma treatment and their diffusion will be suppressed (Figure 9c).

![Figure 6: Impacts of MOCVD TiN growth time on EOT and Vfb.](image6)

![Figure 7: Impacts of MOCVD TiN plasma treatment time on EOT and Vfb.](image7)

![Figure 8: HRTEM images of samples as listed in table.](image8)
accelerate interfacial oxide formation and increase equivalent oxide thickness (EOT) due to oxygen diffusion of MOCVD TiN film into IL layer by the action of plasma treatment.

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