Acid-triggered core cross-linked nanomicelles for targeted drug delivery and magnetic resonance imaging in liver cancer cells

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Purpose: To research the acid-triggered core-crosslinked folate-poly(ethylene glycol)-b-poly[N-(N’,N’-disopropylaminoethyl) glutamine] (folated-PEG-P[GA-DIP]) amphiphilic block copolymer for targeted drug delivery and magnetic resonance imaging (MRI) in liver cancer cells.

Methods: As an appropriate receptor of protons, the N,N-diisopropyl tertiary amine group (DIP) was chosen to conjugate with the side carboxyl groups of poly(ethylene glycol)-b-poly(L-glutamic acid) to obtain PEG-P(GA-DIP) amphiphilic block copolymers. By ultrasonic emulsification, PEG-P(GA-DIP) could be self-assembled to form nanosized micelles loading doxorubicin (DOX) and superparamagnetic iron oxide nanoparticles (SPIONs) in aqueous solution. When PEG-P(GA-DIP) nanomicelles were combined with folic acid, the targeted effect of folated-PEG-P(GA-DIP) nanomicelles was evident in the fluorescence and MRI results.

Results: To further increase the loading efficiency and the cell-uptake of encapsulated drugs (DOX and SPIONs), DIP (pK\textsubscript{a} ≈ 6.3) groups were linked with ∼50% of the side carboxyl groups of poly(L-glutamic acid) (PGA), to generate the core cross-linking under neutral or weakly acidic conditions. Under the acidic condition (eg, endosome/lysosome), the carboxyl groups were neutralized to facilitate disassembly of the P(GA-DIP) blocks’ cross-linking, for duly accelerating the encapsulated drug release. Combined with the tumor-targeting effect of folic acid, specific drug delivery to the liver cancer cells and MRI diagnosis of these cells were greatly enhanced.

Conclusion: Acid-triggered and folate-decorated nanomicelles encapsulating SPIONs and DOX, facilitate the targeted MRI diagnosis and therapeutic effects in tumors.

Keywords: acid-triggered, DOX, nanomicelle carrier, folate-targeted effect, SPIONs

Introduction

Effective targeted nanoparticle encapsulating contrast agents could have a large impact on the future of early MRI tumor diagnosis. As a sensitive and negative contrast agent, superparamagnetic iron oxide nanoparticles (SPIONs) could be encapsulated into some nanomicelles. A previous study indicated that folate-functionalized poly(ethylene glycol)-b-poly(ε-caprolactone) (PEG-PCL) micelles were used for targeted delivery of magnetic resonance imaging (MRI) contrast agents and antitumor drugs. These nanocarriers have attracted great interest, due to their multifunctional characteristics, including the ability to target special cell surface receptors of cancer cells. For example, a previous study reported that the tumor-targeted multifunctional nanomicelles loading SPIONs and doxorubicin (DOX) could be used for MRI diagnosis and targeted cancer therapy.
Polymeric nanocarriers, particularly nanosized micelles capable of passive or active targeted effect, have long been considered an appropriate and reliable delivery system of SPIONs. However, similar to the drug’s encapsulation, the high contrast parameters of SPIONs depend on the stability and loading efficiency of the micelles, which may require the specific and smart modification of the nanocarriers. During the storage and transportation of drug carriers, widely practiced core cross-linking effectively prevents the encapsulated agents from being released and the shell–core structure from decomposing prematurely. Hereinto, covalent cross-linking is an approach that is frequently used to stabilize drug encapsulations. Environmentally, responsive technologies (eg, those based on pH sensitivity, thermal sensitivity, and redox sensitivity), combined with covalent cross-linking, have exhibited outstanding performance in the controlled release of encapsulated drugs. In addition, ionic cross-linking, namely the electrostatic interaction between the positive and negative charges of the hydrophobic polymer blocks, is also a good choice in the micellar core.

On the other hand, weakly acidic interstitial fluid (pH 6.75–7.23) in solid tumors and acidic endosomes/lysosomes (pH 4.0–5.5) in cells possess the characteristic acidic environment. Using these factors, it is possible to construct acid-responsive nanocarriers, whose drug release is triggered in a site-specific manner for malignant tumors. In this process, some acid-sensitive chemical groups which become deprotonated in a neutral environment and protonated in an acidic environment are introduced into the hydrophobic segment of the amphiphilic block copolymers, to build up the drug-loaded nanomicelles. Based on such a structural design, the hydrophobic core of the nanocarriers undergoes a hydrophobic to hydrophilic transition once trapped in the acidic environment (endosome/lysosome), which leads to the swelling or even collapse of the nanoassemblies, and thus triggering the fast release of the encapsulated drugs into the hydrophobic core. In previous reports, modified acid-sensitive poly(acrylic acid) (PAA) and poly(amino acid) commonly were used to construct the site-specific drug nanocarriers. Although PAA can easily be functionalized to respond to external stimuli (such as light, heat, and acid), in vivo application of the drug carrier is greatly limited due to its non-biodegradability. Compared to PAA, poly(amino acid)s such as poly(L-glutamic acid) (PGA), poly(L-aspartic acid), and poly(L-lysine) have better in vivo biocompatibility and biodegradability, and bear reactive side carboxyl or primary amine groups that can easily be modified to confer stimuli-responsive capabilities. Moreover, as an available acid-responsive polypeptide, PGA could undergo a sharp-phase transition by helix–coil conformational change to significantly benefit the accelerative release of the entrapped drug, once the pH value is close to or below its pKa (≤4.25).

In the present study, PGA was chosen as the basic component for the construction of the hydrophobic block of the amphiphilic copolymers in acid-sensitive nanomicelles. To enable acid-sensitivity according to the inter- or intracellular environment of live cancer cells, N,N-diisopropyl tertiary amine group (DIP, pKa = 6.3) was grafted to PGA through amidation. In the neutral aqueous solution, synthesized folate-terminal poly(ethylene glycol)-b-poly[N-(N′,N′-disopropylaminoethyl) glutamine] (folate-PEG-P[GAL-DIP]) was used to produce self-assembled nanomicelles for suitable drug encapsulation and delivery. In a weakly acidic substance (the tumor–intercellular substance), DIP (pKa = 6.3) could accept one proton to convert hydrophobic P(GA-DIP) blocks into hydrophilic electropositive ones. To obtain the ionic cross-linking between the electropositive P(GA-DIP) and the remaining electronegative carboxyl groups of PGA (pKa = 4.25), DIP was grafted to ~50% of the PGA’s carboxyl groups for stability of the nanomicelles. If DIP segments were grafted to 100% carboxyl groups of PGA (without the remaining carboxyl groups), the strong repulsion among the protonated P(GA-DIP) blocks would destabilize and disassemble the nanomicelles, initiating drug release at too high a rate. In an acidic substance (eg, endosome/lysosome), these carboxyl groups of PGA (pKa = 4.25) could be protonated to become electroneutral and hydrophobic components, according to the carboxyl’s properties. In this case, an attraction could not exist between the electropositive P(GA-DIP) segments and the remaining electroneutral carboxyl groups of PGA, thus, terminating the ionic cross-linking and triggering fast drug release.

If the nanomicelles are connected with the tumor-targeted folate-ligands, the encapsulated DOX and SPIONs would effectively be entrapped in the cancer cells for tumor-specific chemotherapy and MRI. The folate receptor (FR) is a valuable therapeutic target that is highly expressed in a variety of cancers, including liver cancer. To further observe the target effect in the present article, the folate-PEG-P(GA-DIP) nanocarriers encapsulating DOX and SPIONs were evaluated in the human liver cell line Bel-7402. Poly(ethylene glycol)-b-poly(D,L-lactic acid) (PEG-PDLLA) and PEG-PCL nanomicelles were used as contrasts.
Materials and methods

Synthesis and characterization of folate-PEG-P(GA-DIP)

According to the synthetic route of folate-PEG-P(GA-DIP) (shown in Figure 1), α-allyl-ω-hydroxy poly(ethylene glycol) (allyl-PEG-OH) (M \(_{n}\) [number average molecular weight]: 1000 g/mol) was synthesized by an anionic ring-opening polymerization of ethylene oxide initiated by potassium naphthalide, as described in earlier reports,\(^{28,29}\) which was followed by the conversion into α-allyl-ω-amino poly(ethylene glycol) (allyl-PEG-NH\(_2\)).\(^{30,31}\) The synthesis and purification of γ-benzyl-L-glutamate N-carboxyanhydride (BLG-NCA) followed the established methodology, and originated from L-glutamic acid (Sinopharm Chemical Reagent Co, Ltd, Shanghai, People’s Republic of China).\(^{32-34}\)

Allyl-poly(ethylene glycol)-b-poly(γ-benzyl L-glutamate) (allyl-PEG-PBLG) was prepared using a standard NCA method.\(^{35,36}\) The removal of the benzyl group in PBLG block was achieved using an acid deprotection method.\(^{37,38}\) After N,N-carbonyl diimidazole (CDI; Sigma-Aldrich, St Louis, MO, USA), activated by N,N-diisopropylamino ethylamine (DIPA; Sigma-Aldrich), was linked to the side carboxyl group of the synthesized allyl-polyethylene glycol-b-poly(L-glutamate) (allyl-PEG-PGA), the terminal allyl group was converted into an amino group on the base of a radical addition reaction of 2-aminoethanethiol hydrochloride, allowing it to link with folic acid (BR; Sinopharm Chemical Reagent Co, Ltd), preactivated by N-hydroxysuccinimide (NHS; Sigma-Aldrich) and dicyclohexylcarbodiimide (Sigma-Aldrich).\(^{28,29}\)

Then the main products were characterized by proton nuclear magnetic resonance (\(^{1}H\)-NMR) spectroscopy (Mercury-Plus 300; Varian Medical Systems Inc, Palo Alto, CA, USA) as shown in Figure 2. For experimental preparation, DIPAEA was redistilled under high vacuum conditions. All analytic grade reagents were redistilled and dried before synthesis. All dialyzers (Shanghai Green Bird Technology Development Co, Ltd, Shanghai, People’s Republic of China) were saved in a 1 mM aqueous ethylene diamine tetraacetic acid (EDTA) solution.

For the contrast groups in the drug release test, the PEG-PDLLA and PEG-PCL diblock copolymers were synthesized by the ring opening polymerizations of the lactone monomer in the presence of a stannous octoate catalyst, and initiated by allyl-PEG-OH (M \(_{n}\); 6000 g/mol) or methoxy-ω-hydroxy polyethylene glycol (M \(_{ω}\); 2000 g/mol), respectively.\(^{39}\) The molecular weights (M \(_{n}\)) of poly(L-lactide) and the PCL block as determined from \(^{1}H\)-NMR were 2000 g/mol and 20,000 g/mol, respectively.

Preparation and characteristics of the DOX and SPION-loaded PEG-P(GA-DIP) nanomicelles

Preparation of the DOX and SPION-loaded PEG-P(GA-DIP) nanomicelles

SPION nanoparticles (6 nm, Fe\(_{3}\)O\(_{4}\) particles) were synthesized according to the previously reported methods.\(^{40}\) Dimethyl sulfoxide (DMSO)/tetrahydrofuran solution (2 mL, volume ratio =1:1) containing 10 mg copolymer, 2 mg hydrophobic DOX (in acid-base titration) (Zhejiang Hisun Pharmaceuticals Co, Ltd, Shanghai, People’s Republic of China) and 2 mg hydrophobic SPION (6 nm Fe\(_{3}\)O\(_{4}\) particles) was added droplet by droplet into 7 mL ultrapure water under ultrasonic action. Then this mixture was dialyzed against water (M \(_{ω}\) cut-off: 1000 Da) for 2 days. The formation of micelles is shown in Figure 3.

Determination of the DOX and SPION-loaded PEG-P(GA-DIP) nanomicelles

The configuration of micelles was revealed using a thermal field emission environmental scanning electron microscope (SEM; Quanta 400; FEI Company, Hillsboro, OR, USA). The particle sizes of the obtained micelles were measured by a zeta potential and particle-size analyzer (Zeta Plus/90Plus; Brookhaven Instruments Corporation, Holtsville, NY, USA).

Figure 1 Synthetic route of allyl-PEG-P(GA-DIP).

Note: 1 represents TsCl and 2 represents NH\(_{3}\)·H\(_{2}\)O.

Abbreviation: allyl-PEG-P(GA-DIP), allyl-poly(ethylene glycol)-b-poly[N-(N′,N′-dipropylaminoethyl) glutamine].
in ultrapure water. The SPION and DOX concentration in the nanomicelles was detected by the following methods. Lyophilized dialyzate (3 mL) was dissolved in 10 mL DMSO/chloroform solution (volume ratio = 1:1). The maximal DOX absorbance was determined over the range 482–502 nm, using an ultraviolet-visible (UV-Vis) spectrophotometer (UV-3150; Shimadzu, Kyoto, Japan) to quantify the loading of DOX in the micelles. Similarly, 1 mL of lyophilized dialyzate was fully dissolved in 5 mL 10% HCl aqueous hydrochloric acid solution to determine the loading of SPION, using an atomic absorption spectrophotometer (Model: Z-2000; Hitachi Ltd, Tokyo, Japan).
Particle sizes of DOX and SPION-loaded micelles at different pH values
Several samples, which included a 5 mL solution of the allyl-PEG-P(GA-DIP) micelles loading DOX and SPION (about 0.56 mg/mL) and 3 mL ultrapure water, were adjusted to various stable pH values (range 3.0–7.0). The particle sizes under different pH environments were measured using a zeta potential and particle size analyzer (ZetaPlus/90Plus).

Drug release from various DOX and SPION-loaded micelles
Various freeze-dried nanomicelles including PEG-PCL, PEG-PDLLA, and PEG-P(GA-DIP) (20 mg each) were resuspended in phosphate-buffered saline (PBS). Each nanomicelle solution was divided equally into two parts and transferred into different dialysis bags (MW cut-off: 14,000 Da). Three dialysis bags (including three nanomicelles in each) were placed in 25 mL PBS solution with a pH of 7.0. The other three dialysis bags were placed in 25 mL PBS solution with a pH of 5.0.

The release study was performed at 37°C in two incubator shakers (ZHWY-200B; Shanghai ZHICHENG Analytical Instrument Manufacturing Co., Ltd, Shanghai, People’s Republic of China). At predetermined time intervals, 5 mL PBS solution outside of the dialysis bag was replaced. The released DOX in the replaced PBS solution could be detected using UV-Vis analysis (480 nm). Fresh PBS solutions of pH 5.0 and 7.0 were added into two shakers, to retain a constant pH environment in the shakers. Then the percentages of the released DOX were calculated cumulatively. Every cumulative release curve for a specific pH value was expressed as an average of two series of tests per sample.

Cell biology experiment
Fluorescence microscopy
The receptors of folic acid were abundant on the surface of human hepatic carcinoma cells (Bel-7402). The cells were purchased from the Experimental Animal Center of Sun Yat-Sen University (Guangzhou, People’s Republic of China). A folate-free Roswell Park Memorial Institute (RPMI) 1640 medium (Life Technologies, Carlsbad, CA, USA) was used, which was supplemented with 10% heat-inactivated fetal bovine serum.

To indicate the successful encapsulation of DOX into folate-PEG-P(GA-DIP) nanomicelles, Bel-7402 cells (1×10⁵) were mixed with the PBS and folate-PEG-P(GA-DIP) (DOX concentration of 0.19 and 0.76 µg/mL, respectively). The nanomicelles were not added into the cells of the PBS contrast group.

In the second test, Bel-7402 cells (1×10⁵) were mixed with the folate-targeted folate-PEG-P(GA-DIP) nanomicelles in the targeted group (0.76 µg/mL DOX concentration). In the non-targeted group, the cells were mixed with the non-targeted PEG-P(GA-DIP) nanomicelles (0.76 µg/mL DOX concentration). In the competitive inhibition group, cells were incubated with folate-PEG-P(GA-DIP) nanomicelles (0.76 µg/mL DOX concentration) and 1 mM free folic acid.

After 2 hours of incubation, the cells were washed three times with 0.5 mL PBS to remove the non-ingested and free nanomicelles. Fluorescence microscopy (A1, Axios Observer; Carl Zeiss Microscopy Gmbh, Jena, Germany) was used to observe the intracellular DOX fluorescence at 490 nm. Then the relative fluorescence uptake was obtained by a
flow cytometer (FACSCalibur; Becton Dickinson, San Jose, CA, USA). The free DOX solution was incubated with cells in the contrast group. Note that the free DOX solution and DOX-loaded nanomicelles had the same DOX concentration. Finally, the relative fluorescence uptake was calculated by comparing the cell DOX fluorescence of the test groups to that of the contrast group.

In vitro MRI scan
After 24 hours of incubation in a humidified incubator (5% CO₂) at 37°C, Bel-7402 cells (5×10⁶) were mixed with folate-PEG-P(GA-DIP) nanomicelles at different iron (Fe) concentrations (0.358, 0.179, 0.0895, and 0.04475 µg/mL). After 2 hours of incubation, the cells were washed three times with 1 mL PBS, and then digested using pancreatic enzyme. The cell solutions were centrifuged, and the supernatant liquid was removed. Then the cell sediment was mixed with 150 µL gelatin (4%) to suspend the cell sediment again. The cell suspension was scanned by a 1.5T MRI scanner (Philips Intera 1.5T; Philips, Amsterdam, Netherlands). PBS without cell sediment was used as the blank contrast. In the gelatin control group, cells were not incubated with nanomicelles and the cell sediment was mixed with gelatin to maintain the stability of the cell suspension.

In the second test, Bel-7402 cells (5×10⁶) were mixed with the folate-targeted folate-PEG-P(GA-DIP) nanomicelles and the non-targeted PEG-P(GA-DIP) nanomicelles at 0.179 µg/mL Fe concentration. In the competitive inhibition group, cells were incubated with folate-PEG-P(GA-DIP) nanomicelles using a combination of 0.179 µg/mL Fe concentration and 1 mM free folic acid together. Folic acid competed with the folate-PEG-P(GA-DIP) nanomicelles for folate-receptors. In the contrast group, the nanomicelles were not added into the cells and the cell sediment was mixed with gelatin. After 2 hours of incubation, the cell suspensions were taken and scanned.

A circular surface coil with an inner diameter of 11 cm was used to take the T2-weighted images and T2-mapping images. The T2-weighted images were acquired using the following parameters: repetition time/echo time (TR/TE) of 2600 ms/100 ms; field of view (FOV) of 150 mm; matrix of 512×512; and slice thickness of 0.6 mm. The T2-mapping images were acquired using the following parameters: SE 8-echo sequence; TR/TE of 1300 ms/(10–90) ms; FOV of 150 mm; matrix of 512×512; and slice thickness of 0.6 mm. The T2-mapping images were color-coded: blue represented a low value and red represented a high value. Then the T2 values of different groups were calculated.

Results and discussion
Synthesis of folate-PEG-P(GA-DIP)
Due to their steric stabilization, biocompatibility, and excellent solubility in aqueous media, the PEG-based macroinitiators have been widely used for preparing diblock and triblock copolymers. As shown in Figure 1, the obtained allyl-PEG-NH₂ was employed as an initiator to synthesize allyl-PEG-PBLG by the ring-opening polymerization of BLG-NCA. After deprotection of the benzyl group, the activated DIP was connected to the side carboxyl group of the PGA block in the allyl-PEG-PGA copolymer.

Figure 2 shows the 1H-NMR spectra of the intermediate product (PEG-PBLG, PEG-PA) and the final PEG-P(GA-DIP) nanomicelles. The –CH₂– peaks in the repeated PEG blocks of the intermediate and final products were close to 3.5 ppm in all the 1H-NMR spectra. For instance, the –CH₂– peak in PEG-PBLG was at 3.48 ppm (“g” peak), the peak in PEG-PGA was at 3.48 ppm (“e” peak), and the peak in PEG-P(GA-DIP) was at 3.54 ppm (“i” peak). In the spectra of PEG-PBLG, the “f” peak at 7.19 ppm represented C₄H₄ in the PBLG block, indicating successful synthesis of PEG-PBLG. In the spectra of PEG-PGA, the peak at 7.19 ppm was not clearly observed, indicating the removal of the benzyl group in PEG-PBLG and successful synthesis of PEG-PGA.

The “h” peak at ~1.0 ppm reflected the CH₃– signal in DIP and indicated successful synthesis of PEG-P(GA-DIP). Furthermore, the PEG-P(GA-DIP) structure was calculated by comparing the integral ratios of different characteristic peaks (listed in Table 1). These peaks included –CH₂– in PEG (“i” peak at ~3.5 ppm), the first –CH₂– of side chain in PEG (“d” peak at 2.26 ppm), and CH₃– in DIP (“h” peak at ~1.0 ppm). More importantly, the grafting percentage of the DIP groups was 48.15%, very close to 50%, similar to the previously determined value. This indicated that ~50% of the carboxyl groups of PGA were grafted with DIP, with ~50% of the PGA’s carboxyl groups remaining.

The three peaks at 6.6 ppm, 7.6 ppm, and 8.6 ppm) shown in Figure 2 (the 1H-NMR spectra of folate-PEG-P(GA-DIP)) indicated the presence of the folate-ligand in the folate-PEG-P(GA-DIP) and successful linkage between the folate and the terminal amino of the PEG block after 5 days of dialysis.

Characteristics of the DOX and SPION-loaded nanomicelles
Detection of the DOX and SPION-loaded PEG-P(GA-DIP) nanomicelles
SEM images indicated the existence of spherical particles (∼60 nm in diameter), which suggested the formation of
allyl-PEG-P(GA-DIP) micelles (Figure 4). Successful detection of DOX and SPIONs (Table 1) allowed for the encapsulation of DOX and SPIONs in the nanomicelles. The encapsulation of DOX and SPIONs was also verified by the tests of MRI and fluorescence.

In general, the nanomicelles with a longer hydrophobic block had the larger self-assembled micelles and higher drug loading. In the present research, PEG-PCL and PEG-PDLLA were also prepared for the contrast groups. Hence, the PEG-PCL micelle loading DOX and SPIONs had the largest particle size (226.2 nm) and highest encapsulation percentage of DOX and SPIONs (42.7% and 14.1%, respectively) among all nanomicelles in Table 1. Interestingly, the allyl-PEG-P(GA-DIP) micelle exhibited a similar encapsulation percentage of DOX, close to that of PEG-PCL (39.7% versus 42.7%). And the DOX loading efficiency of the allyl-PEG-P(GA-DIP) micelle was higher than that of PEG-PCL (16% versus 8.1%). In a neutral aqueous solution, one primary amino group of partial DOX (pK1 = 8.6) was protonated and bound with the remaining carboxyl groups of PGA (pK1 = 4.25). An attraction existed between the electropositive DOX and the remaining electronegative carboxyl groups of PGA. Hence, we chose only 50% of the PGA’s carboxyl groups to connect with the DIP, to achieve a high encapsulation percentage and loading efficiency of DOX. If the grafting between DIP and the carboxyl groups of PGA reached 100%, then the remaining carboxyl groups disappeared and the DOX loading decreased. Of course, the above situation did not exist in the encapsulation percentage of acid-covered SPIONs. The encapsulation percentage of allyl-PEG-P(GA-DIP) (7.8%) was lower than that of PEG-PCL (14.1%).

In addition, the DOX and SPION encapsulation percentages of folate-PEG-P(GA-DIP) (3.2% and 5.5%, respectively) were lower than that of allyl-PEG-P(GA-DIP). During the formation of folate-PEG-P(GA-DIP) micelles, the primary amino groups of the outer folic acid molecules partially inhibited DOX from embedding itself into the P(GA-DIP) core, which resulted in a low DOX encapsulation percentage and loading efficiency, different from the allyl-PEG-P(GA-DIP) micelles.

Particle sizes of the drug-loaded PEG-P(GA-DIP) micelles at different pH values
Figure 5 shows the particle sizes of the drug-loaded allyl-PEG-P(GA-DIP) micelles in various pH environments.

Table 1 Chemical structure and carrier performances of allyl-PEG-P(GA-DIP), folate-PEG-P(GA-DIP), and contrast copolymers

| Sample | Allyl-PEG-P(GA-DIP) | Folate-PEG-P(GA-DIP) | PEG-PDLLA | PEG-PCL |
|--------|---------------------|----------------------|-----------|---------|
| Molecular weight of PEG block (g/mol) | 1000 | 1000 | 6000 | 6000 |
| Amount of repeating units of PBLG | 2 | 2 | | |
| Amount of repeating units of PGA | 12 | 12 | | |
| Amount of repeating units of P(GA-DIP) | 13 | 13 | | |
| Molecular weight of hydrophobic block (g/mol) | 6700 | 6700 | 2000 | 20,000 |
| Total molecular weight (g/mol) | 7700 | 7700 | 8000 | 22,000 |
| Grafting percent of DIP (%) | 48.2 | 48.2 | | |
| Loading efficiency of DOX (%) | 16.0±3.3 | 1.1±0.2 | 6.3±0.8 | 8.1±1.9 |
| Encapsulation percent of DOX (%) | 39.7±6.9 | 3.2±2.1 | 20.2±1.6 | 42.7±9.8 |
| Loading efficiency of SPIONs (%) | 2.2±0.6 | 2.2±1.2 | 5.6±1.1 | 9.3±1.0 |
| Encapsulation percent of SPIONs (%) | 7.8±0.2 | 5.5±0.4 | 6.6±2.9 | 14.1±1.5 |
| Particle size (nm) | 58.9±0.4 | 65.8±6.9 | 53.5±3.0 | 226.2±3.2 |

Notes: All data about chemical structure were calculated by relative 1H-NMR spectrum. *Expressed as the averaging of at least three results.

Abbreviations: DIP, N,N-diisopropyl tertiary amine group; DOX, doxorubicin; 1H-NMR, proton nuclear magnetic resonance; PBLG, b-poly(γ-benzyl L-glutamate); PEG-P(GA-DIP), poly(ethylene glycol)-b-poly[N-(N′,N′-diisopropylaminoethyl) glutamine]; PGA, poly(L-glutamic acid); SPIONs, superparamagnetic iron oxide nanoparticles; PEG-PDLLA, poly(ethylene glycol)-b-poly(D,L-lactic acid); PEG-PCL, poly(ethylene glycol)-b-poly(e-caprolactone).
As we can see, all of the micelles’ particle sizes also slightly decreased from 64.2 to 59.9 nm except the one at pH 6.0, while the pH value of the micelle solution decreased from 7.0 to 4.0. In this process, the charge property of some acid-responsive groups would vary with the concentration of hydrogen protons (H⁺). For example, DIP groups would be protonated to be more and more hydrophilic, and electropositive, with the decrease of pH values (from pH 10.0 to 5.5). This was a process of dynamic balance, especially in view of the existence of PGA’s side carboxyl groups with the opposite charge. At pH 6.0, DIP groups could not be fully protonized, which was not enough to neutralize the residual negative side carboxyl groups of PGA (pKₐ = 4.25). The amount of positive and negative charges could not keep equilibrium at pH 6.0, but could do so at pH 4.0. So these acid-sensitive micelles were enlarged at pH 6.0 and were smallest at pH 4.0.

Importantly, all micelles generating positive charges of DIP would also form the strong electric attraction with the PGA’s residual negative carboxyl groups (from pH 6.0 to 4.0), which built up the core cross-linking structure and micellar stability. In this process, the acid-sensitivity of DIP could be continuously electropositive. The electric attraction could strengthen between the electropositive DIP groups and the remaining electronegative carboxyl groups of PGA (pKₐ = 4.25), because 50% of the carboxyl groups of PGA were not grafted with the DIP.

The particle size of the nanomicelles had an outstanding enlargement at a pH of 3.0. As this pH value was significantly lower than the PGA’s pKₐ (=4.25), all of the remaining carboxyl groups of PGA would be completely electroneutral according to the carboxyl groups’ individual properties. In this case, the electric attraction and cross-linking between the P(GA-DIP) groups and the remaining carboxyl groups would disappear. In contrast, the mutual repulsion had an obvious effect inside the protonated electropositive DIP groups at a pH of 3.0. Hence, the micelles quickly expanded to 103.1 nm.

Figure 3 describes the size and change in cross-linking of acid-triggered nanomicelles at different pH environments.

Drug release from various drug-loaded nanomicelles
The release of DOX from various DOX-loaded nanomicelles was studied at a pH of 5.0 or 7.0 (Figure 6). At a pH of 7.0, less than 20% of the DOX in all of the nanomicelles was released after 30 days. All of the nanomicelles demonstrated slow DOX release in the neutral environment.

However, at a pH of 5.0, DOX release in the allyl-PEG-P(GA-DIP) micelles was greatly accelerated. After 2 days, 41.5% of the DOX was released from the allyl-PEG-P(GA-DIP) micelles. Only 11.4% and 10.3% of the DOX in PEG-PDLLA and PEG-PCL micelles were released, respectively. After 27 days, half of the DOX was released from the allyl-PEG-P(GA-DIP) micelles at a pH of 5.0, and 40.2% and 15.5% of the DOX in PEG-PDLLA and PEG-PCL micelles were ultimately released, respectively.

For PEG-PDLLA, its good degradability and short hydrophobic blocks (only 2000 g/mol) facilitated rapid micelle disassembly, catalyzed by high acid concentrations. For allyl-PEG-P(GA-DIP), a pH of 5.0 was much lower than the DIP’s pKₐ (=6.3), so that the P(GA-DIP) groups could be electropositive and hydrophilic. At a pH of 5.0, it was close to the pKₐ of the PGA’s remaining carboxyl groups (pKₐ = 4.25), and the partially remaining carboxyl groups became weakly electronegative or even electroneutral according to their individual properties. Under these conditions, the cross-linking between the electropositive P(GA-DIP) groups and the remaining carboxyl groups (electronegative or even electroneutral) decreased. Moreover, the hydrophobic P(GA-DIP) blocks changed into the hydrophilic ones. Due to the decreased cross-linking and the increased hydrophilic P(GA-DIP), the encapsulated DOX could be released quickly. This hydrophilic transformation of the P(GA-DIP) core occurred earlier than the visible change in its size. It is possible that DOX could be also released quickly in endosomes and lysosomes, where the pH environment was close to or slightly lower than 5.0.

Specific and fast drug release from acid-triggered nanomicelles was very effective inside the cancer cells. In some
previous reports, the pH-catalyzed hydrolysis of drug-polymer linkages in the polymer drugs had rapid drug release rates and high drug efficacy.43,44 The lysosomal membrane could hinder the location of the ingested nanomicelles. The encapsulated drugs in the nanomicelles were mainly released in the lysosome. Hence, the small size of the released drugs easily permeated the lysosomal membrane to reach the targeted site.

Cell biology experiment

Fluorescence microscopy

In the cell-uptake test of nanomicelles (Figure 7, 100×), the red-colored DOX fluorescence could be seen in the micelle groups (DOX concentration: 0.19 and 0.76 µg/mL), indicating the successful DOX encapsulation in the cells. The fluorescence intensity became strong, while the DOX concentration increased. This also confirmed that DOX encapsulated in micelles entered the cancer cell.

In the targeted test (Figure 8, 100×), the red-colored DOX fluorescence of the targeted group was obvious. The relative fluorescence intensity of the folate-targeted group (86.8%) was higher than that of other groups (Figure 9). In the competitive inhibition and the non-targeted groups, a weak DOX fluorescence was observed. The weak relative fluorescence intensity of the non-targeted group was 56.1%, which was lower than that of the targeted group. Particularly, the free folic acid in the competitive inhibition group was also associated with a folic-receptor, which inhibited further

![Figure 6](image1.png)

Figure 6 Drug release from various nanomicelles at different pH.

**Abbreviations:** allyl-PEG-P(GA-DIP), allyl-poly(ethylene glycol)-b-poly[N-(N′,N′-disopropylaminoethyl) glutamine]; PEG-PCL, poly(ethylene glycol)-b-poly(ε-caprolactone); PEG-PDLLA, poly(ethylene glycol)-b-poly(D,L-lactic acid).

![Figure 7](image2.png)

Figure 7 Fluorescence microscope images.

**Notes:** Bel-7402 cells incubated with PBS and folate-PEG-P(GA-DIP) at different DOX concentrations (µg/mL) for 2 hours. (A) PBS control; (B) 0.19 µg/mL; (C) 0.76 µg/mL. **Abbreviations:** DOX, doxorubicin; folate-PEG-P(GA-DIP), folate-poly(ethylene glycol)-b-poly[N-(N′,N′-disopropylaminoethyl) glutamine]; PBS, phosphate-buffered saline.
connections between targeted micelles and receptors to generate the low relative fluorescence intensity (45.5%). The folate-targeted effect enhanced the cell uptake of folate-PEG-P(GA-DIP) nanomicelles loading DOX, showing that the linkage between folic acid and the acid-responsive PEG-P(GA-DIP) copolymer worked well in liver cancer cells.

In vitro MRI scans

Compared to the PBS control without cells, the cells suspended in the gelatin control group had a lower signal. The gelatin was used to keep the cell suspension stable. In the uptake test (Figure 10), the T2 signal intensity decreased, while the Fe concentration increased. SPIONs were successfully encapsulated into the folate-PEG-P(GA-DIP) nanomicelles. Hence, nanomicelles loading SPIONs were indeed transported into the cells. The drug concentration in the nanomicelles impacted the signal intensity. A low signal was clearly observed for the 0.179 µg/mL Fe concentration group. Hence, this Fe concentration group was chosen for the next target test. In Figure 11, T2 values of several different groups were calculated.
In the targeted test, the signal of the folate-targeted group was lower than that of the non-targeted group and the competitive inhibition group (Figures 12 and 13). The targeted effect was attributed to the high affinity of folate-receptor mediated endocytosis.\textsuperscript{45,46} The PEG-P(GA-DIP) nanomicelles did not have the folate-ligands. Some PEG-P(GA-DIP) nanomicelles were transported into the cells by cell endocytosis. In the competitive inhibition group, the free folic acid decreased the linkage between the folate-targeted nanomicelles and the folate receptors of the cells.

The folate-targeted effect existed in both the primary tumors and metastases. A previous study indicated that folate receptors could be expressed significantly in the squamous cell carcinoma of the head and neck, both in primary tumors and in the corresponding lymph node metastases.\textsuperscript{47} The high expression of folate receptors appeared to be correlated with the clinical outcomes. Folate receptors in metastatic lymph nodes suggested a strong potential for targeted chemotherapy in both primary tumor and metastasis. If a tiny metastasis cannot be detected by normal imaging methods, this targeted effect in metastasis could be considered. The folate-targeted nanomicelles could be used to treat the tiny metastasis and prevent tumor recurrence.

**Conclusion**

On the basis of the grafted DIP groups, the PEG-P(GA-DIP) block copolymer demonstrated good acid responsiveness and improved cellular uptake of encapsulated DOX and SPIONs for effective cancer therapy and diagnosis. Moreover, the linkage of folic acid with PEG-P(GA-DIP) further facilitated drug delivery by nanomicelles to targeted cancer cells.

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**Disclosure**

The authors report no conflicts of interest in this work.

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