Valence Band Splitting in Wurtzite InP Nanowires Observed by Photoluminescence and Photoluminescence Excitation Spectroscopy

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ABSTRACT
We have investigated individual bulk-like wires of wurtzite InP using photoluminescence, photoluminescence excitation spectroscopy and transmission electron microscopy. Using two different methods we find that the top of the valence band is split, as expected theoretically. This splitting of the valence band is peculiar to wurtzite InP and does not occur in zinc blende InP. We find the energy difference between the two bands to be 40 meV.

KEYWORDS
wurtzite InP, photoluminescence, excitation spectroscopy, valence band structure

Recently, the possibility of controlling the crystal phase in nanowires (NWs)—which opens up new opportunities for band structure engineering using only a single material—has received much interest. In particular, the crystal structure of InP NWs can be controlled by varying the growth parameters to produce either zinc blende (ZB) or wurtzite (WZ), as has been shown experimentally [1–7] and modelled theoretically [8–10]. The nanowire diameter [4, 5, 7, 9, 10] in particular is correlated with the crystal structure, where a thinner wire tends to favour wurtzite formation. In addition, in situ nanowire doping has been found to strongly affect the crystal phase [4]. We observed that sulfur doping favours wurtzite structure formation. It is of great interest to characterise the mostly unknown bulk properties of these new types of materials.

Optical investigations at 4 K on wurtzite InP nanowires indicated a bandgap energy of 1.49–1.50 eV [2, 6, 12] which is higher than the bandgap of zinc blende InP (1.41 eV). In order to assess the bulk-like properties of wurtzite InP, we grew 15 nm sulfur-doped InP nanowire cores in the wurtzite phase, as confirmed by transmission electron microscopy. Bulk-like material was then created by growth of a thick intrinsic epitaxial InP shell around the nanowires. By “bulk-like” we mean that the wires were quite thick and no quantum confinement was present.

In this article we report photoluminescence (PL) and PL excitation (PLE) spectroscopy studies of such individual bulk-like wurtzite InP wires. We find that the bandgap energy is in agreement with previous studies. Figure 1 shows the band structure of wurtzite
Figure 1  \(E-k\) diagrams for wurtzite (WZ) and zinc blende (ZB) InP. In wurtzite InP the top of the valence band is split into two bands, each of which is doubly degenerate (where spin is included). The spin-orbit split-off band is also included in the figure. The top of the valence band is fourfold degenerate for zinc blende InP and zinc blende InP: note that the top of the valence band is split in wurtzite InP in contrast to zinc blende InP. We experimentally determine this splitting to be about 40 meV, which is in good agreement with theoretical calculations [11] and PLE experiments [12]. We find that we can populate the second valence band using a high excitation power density, which allows a fairly easy determination of the energy splitting, in comparison with previous work where PLE was needed to determine this parameter [12].

Samples for growth were prepared by depositing 15 nm aerosol particles on a (111)B InP substrate via an aerosol technique [13] resulting in randomly distributed particles with a homogeneous surface density of \(5 \times 10^7/\text{cm}^2\). We used low pressure (100 mbar) metal–organic vapour phase epitaxy (MOVPE) for particle-assisted nanowire growth of the core. Trimethylindium (TMI), phosphine (PH\(_3\)) and hydrogen sulphide (H\(_2\)S) were used as precursors in a total flow of 6 L/min using hydrogen as a carrier gas. HCl was used as an \textit{in situ} etching agent in order to prevent any tapering. The PH\(_3\), TMI, and H\(_2\)S molar fractions (\(\chi(\text{PH}_3)\), \(\chi(\text{TMI})\), \(\chi(\text{H}_2\text{S})\)) were set to \(\chi(\text{PH}_3) = 6.2 \times 10^{-3}\), \(\chi(\text{TMI}) = 3.5 \times 10^{-6}\) and \(\chi(\text{H}_2\text{S}) = 7.1 \times 10^{-6}\) for growth of the wurtzite cores. The HCl molar fraction was set to \(\chi(\text{HCl}) = 8.3 \times 10^{-6}\). The samples were first heated to 550 °C under a PH\(_3\)/H\(_2\) gas mixture for 10 min in order to desorb any surface oxides. The reactor temperature was then reduced to 420 °C, at which temperature growth was initiated by adding TMI to the flow. H\(_2\)S and HCl were added simultaneously after a 15 s nucleation step. After growth of the core for 20 min, TMI, H\(_2\)S, and HCl were switched off simultaneously, and the temperature ramped up to 600 °C, at which point TMI was switched on to grow a non-intentionally doped epitaxial shell for 15 min, resulting in wires with a diameter of 100 nm. Samples were prepared for high-resolution transmission electron microscopy (HRTEM) by direct transfer of the wires to a lacey carbon coated copper grid by gently pressing the grid onto the substrate. The wires were characterised by HRTEM using a JEOL 3000F instrument operating at 300 kV. The wires were tilted into the \(<1\bar{1}20>\) zone axis and imaged over their entire length.

For PL(E) measurements, the wires were deposited on a silicon substrate, covered with a gold layer sufficiently thick to suppress interference from the silicon photoluminescence. PL measurements were carried out using the 532 nm line from a frequency-doubled Nd-YAG laser, which was focused through a microscope objective. The signal was then collected through the same microscope objective, dispersed through a spectrometer and detected by a CCD-camera. For PLE and temperature-dependent measurements a CW Ti:sapphire laser was coupled in at a 45° angle outside the microscope to create a large excitation area with an excitation power density of approximately 3 nW/\(\mu\text{m}^2\). A neutral density filter was used to control the laser power incident on the nanowires. Figure 2 shows a TEM image of a wurtzite InP core shell wire.

![TEM images of InP wurtzite wires. (a) Overview image of one wire viewed along the \(<1\bar{1}20>\) axis. An adjacent wire which is off axis is also shown. The dark bands in the latter indicate the presence of stacking faults and/or short zinc blende segments. (b) A high resolution image showing the pure wurtzite structure which constitutes the majority of the wire.](image-url)
Long segments (> 1 μm) of pure wurtzite structure, which are interrupted by stacking faults or short zinc blende segments, are observed in the wires.

From TEM imaging, the core shell volume ratio is estimated to be 1:50. Sulfur is not expected to diffuse into the shell, due to its low diffusion constant [14]. Therefore, we argue that luminescence from the core–shell wires mainly originates from the large volume of the bulk-like, undoped, InP wurtzite shell. In Fig. 3 we show luminescence from a wire grown without a shell, which shows emission at a fairly high energy, 1.6 eV, representing quite a marked blue shift of the emission energy compared with undoped InP. This is due to the high density of donors. This blue shift is not seen in the wires grown with an undoped shell.

The PL spectra of two InP wurtzite wires, having an undoped shell outside the core are depicted in Fig. 4. The spectra were obtained at 4 K using a maximum excitation power density of $P = 0.1 \text{ mW/μm}^2$. At a low excitation power density, 0.3–3 nW/μm², two peaks can be seen. The higher energy peak at 1.493 eV is attributed to transitions from the conduction band to the valence band, most likely involving excitons. We will refer to this peak as the exciton peak. The exciton binding energy is expected to be low, about 5 meV, and will not substantially affect our results with regards to, for example, the bandgap energy. Our bandgap energy is slightly lower than that measured in previous work [12]. Carrier–carrier interaction is known to reduce the bandgap in some cases [14], but will also destroy the exciton and give broad emission lines, in contrast to our observation. The core itself has a very different emission, see Fig. 3. It is more likely that the

![Figure 3](image1.png) ![Figure 3](image2.png)
small difference in the measured bandgap is due to different band bending in our case, compared with other samples [12]. Some influence of the surface, such as surface charges affecting the band bending, is expected in small wires. The transition at lower energy, 1.488 eV, is attributed to impurities and will be referred to as the impurity peak. We have indicated the bandgap energy by a dotted line. A large number of wires were characterised at an excitation power density of 3 nW/μm² with similar results. No transitions at energies above 1.49 eV were found in any of the wires, at these excitation power densities. The impurity peak varied in intensity relative to the exciton peak depending on the wire (compare Figs. 4(a) and 4(b)) and had an energy about 5–6 meV below the exciton peak. At higher excitation power density, the exciton recombination eventually dominates over the impurity peak, as expected.

At even higher excitation power densities (Fig. 4, colored traces) a peak at higher energy emerges, at an energy around 1.56 eV. This peak, which we observe in every wire, we attribute to a transition between the conduction band and the second valence band (see Fig. 1). Band structure calculations by De and Pryor [11] find the split between the first and the second valence band to be about 60 meV in wurtzite InP, which is in reasonable agreement with our experimental value of about 40 meV. We attribute the tail on the low energy side, at high excitation power density, to bandgap renormalization [2, 15].

In order to strengthen our assignment of this peak to the second valence band, we performed low temperature PLE measurements, shown in Fig. 4. Detecting at 1.48 eV, we observed a PLE onset at an energy around 1.488 eV, which coincides with the impurity-related peak. This onset levels out at an energy around the exciton-related peak, indicated by a dotted line at 1.493 eV. About 40 meV higher, a second onset is visible when the detection is at the exciton peak. This second onset is precisely what is expected if the top of the valence band is split, and confirms our interpretation from the band-filling experiment.

The same energy splitting was also seen in room temperature PL spectra of four different wires (see Fig. 5). The second valence band recombination is indicated by a peak around 1.464 eV, approximately 40 meV higher than the bandgap at 1.427 eV.

In order to gain more insight into the exciton- and impurity-related peaks, temperature-dependent PL measurements were performed (Fig. 5). For these measurements a low excitation power density was used (3 nW/μm²) which results in well separated peaks. The activation energy of the impurity peak is estimated to be around 5 meV, which is close to the energy difference between the impurity peak and the exciton peak observed in photoluminescence. We find that the exciton peak has a relatively constant intensity up to about 40 K and then starts to quench. The total PL intensity decreases as we increase the temperature, and there must thus be non-radiative channels that become more prominent at higher temperatures.
In conclusion, we have measured PL and PLE spectra for pure wurtzite InP wires. From the spectra, the bandgap can be estimated to be about 1.493 eV. We also observe transitions involving a second valence band, present only in wurtzite InP and not in zinc blende InP. We find that the top of the valence band in wurtzite InP is split by about 40 meV, in agreement with other reports [12]. Temperature-dependent PL measurements reveal an impurity peak with a low activation energy, which is always present in the wires, as well as the existence of non-radiative recombination channels.

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