Urinary Excretion of N-Nitroso Compounds in Rats Fed Sodium Nitrite and/or Hot Dogs

Lin Zhou,† Muhammad M. Anwar,‡ Muhammad Zahid,§,† Valerie Shostrom,∥ and Sidney S. Mirvish†,§,¶

†Eppley Institute for Research in Cancer, ‡Department of Environmental, Agricultural, and Occupational Health, College of Public Health, §Department of Pharmaceutical Sciences, and ∥Department of Statistics, University of Nebraska Medical Center, Omaha, Nebraska 68198, United States

ABSTRACT: Nitrite-treated meat is a reported risk factor for colon cancer. Mice that ingested sodium nitrite (NaNO2) or hot dogs (a nitrite-treated product) showed increased fecal excretion of apparent N-nitroso compounds (ANC). Here, we investigated for the first time whether rats excrete increased amounts of ANC in their urine after they are fed NaNO2 and/or hot dogs. Rats were treated for 7 days with NaNO2 in drinking water or were fed hot dogs. Their 24 h urine samples were analyzed for ANC by thermal energy analysis on days 1–4 after nitrite or hot dog treatment was stopped. For two rats fed 480 mg NaNO2/L drinking water, mean urinary ANC excretion on days 1–4 was 30, 5.2, 2.5, and 0.8 nmol/day, respectively. For two to eight rats/dose given varied NaNO2 doses, mean urinary ANC output on day 1 increased from 0.9 (for no nitrite) to 37 (for 1000 mg NaNO2/L drinking water) nmol ANC/day. Urine samples of four rats fed 40–60% hot dogs contained 12–13 nmol ANC on day 1. Linear regression analysis showed highly significant correlations between urinary ANC excretion on day 1 after stopping treatment and varied (a) NaNO2 level in drinking water for rats fed semipurified or commercials diet and (b) hot dog levels in the diet. Some correlations remained significant up to 4 days after nitrite treatment was stopped. Urinary output of ANC precursors (compounds that yield ANC after mild nitrosation) for rats fed semipurified or commercial diet was 11–17 or 23–48 μmol/day, respectively. Nitrosothiols and iron nitrosyls were not detected in urinary ANC and ANCP. Excretion of urinary ANC was about 60% of fecal ANC excretion for 1 to 2 days after NaNO2 was fed. Administered NaNO2 was not excreted unchanged in rat urine. We conclude that urinary ANC excretion in humans could usefully be surveyed to indicate exposure to N-nitroso compounds.

1. INTRODUCTION

We report here studies on the urinary excretion by rats of total apparent N-nitroso compounds (ANC) formed in vivo from ingested sodium nitrite (NaNO2) or hot dogs. This appears to be the first published report on the urinary excretion of ANC by rodents. A significant proportion of ingested N-nitroso compounds (NOC) appears to arise from processed nitrite-treated meat. Most of the ANC in nitrite-treated meat products appears to consist of NOC. Fresh and, especially, processed (mostly nitrite-treated) red meat products, including hot dogs, are reported risk factors for the etiology of colon cancer. In 1991, Rowland et al. proposed that NOC in red meat could induce colon cancer if these NOC reached the colon. In fact, ANC were detected in the feces of humans and mice after they ingested fresh or processed red meat.

Processed meat is also a reported risk factor for the etiology of human cancer of several organs in addition to the colon. These organs include the prostate, pancreas, breast, esophagus, and brain. NOC exposure in these organs could be detected by measuring urinary rather than fecal excretion of NOC. Linkages with cancer of the gastrointestinal (GI) tract could occur if GI tissues directly absorbed carcinogenic NOC from the GI lumen.

Santorelli et al. reported experiments indicating that (a) ANC and oxidized lipids are the agents in processed meat that induce colon cancer and (b) these effects were exacerbated by also feeding hemoglobin. In later studies, rats were injected with 1,2-dimethylhydrazine (a colon carcinogen) and then fed hot dogs. They showed more colonic mucin-depleted foci (a precursor lesion for colon cancer) and higher levels of fecal ANC than did rats receiving only dimethylhydrazine. When other similarly treated rats were also fed calcium carbonate or α-tocopherol, these effects were significantly reduced.

Here, we investigated the effect of feeding NaNO2 on urinary ANC excretion by rats because NaNO2 is the major additive in processed meat and feeding only 32 mg NaNO2/L significantly increased fecal ANC excretion in mice. This effect was attributed largely to acid-catalyzed gastric nitrosation of NOC precursors (NOCP, usually measured as apparent NOCP (ANCP)). Nitrosated ANCP that were partially purified from hot dogs were directly mutagenic in the Ames test on bacteria (a property well correlated with rodent carcinogenicity) and induced colonic aberrant crypts in mice. (This is another putative precursor of colon cancer.) These findings support the
hypothesis that NOC present in or derived from NOCP in processed meat are a cause of colon cancer.

In addition to NOC, ANC can include nitrosothiols (RSNO) and nitrosyl iron compounds (RFeNO), which are probably not carcinogenic because they are unlikely to alkylate DNA bases. Therefore, we analyzed some urine samples for RSNO and RFeNO in addition to total ANC. We studied rats here, rather than mice as we had done before, because the larger urine volume for rats made it easier to collect their urine. The rats were mostly fed semipurified diet. Commercial diet was also fed in some tests because it is similar to most human diets.

2. METHODS

2.1. Treatment of Rats and Collection of Urine and Feces. Animal experiments were approved by the Institutional Animal Care and Use Committee of the University of Nebraska Medical Center. The study animals were male Sprague-Dawley rats (Charles Rivers, Wilmington, MA) aged 8–12 weeks and housed two/cage. They weighed 250–350 g and excreted 6–10 mL urine/day. The rats were fed tap water for drinking (distilled water sometimes contained low levels of nitrite, probably derived from atmospheric nitrogen oxides) or tap water containing 60, 120, 240, 480, or 1000 mg/L of NaNO2 (Sigma-Aldrich Reagent Plus grade). The diet was pelleted AIN93G diet (TD-94245) or pelleted commercial diet (TD-98186), both from Harlan Teklad (Madison, WI).

In other experiments, 12 rats were fed Bar-S hot dogs (smoked sausages, Bar-S Foods, Phoenix, AZ) containing, in order of abundance 98186), both from Harlan Teklad (Madison, WI). Both hot dog brands were manufactured with the addition of NaNO2. The remainder of these animals were male Sprague–Dawley rats (Charles Rivers, Wilmington, MA) aged 8–12 weeks and housed two/cage. They weighed 250–350 g and excreted 6–10 mL urine/day. The rats were fed tap water for drinking (distilled water sometimes contained low levels of nitrite, probably derived from atmospheric nitrogen oxides) or tap water containing 60, 120, 240, 480, or 1000 mg/L of NaNO2 (Sigma-Aldrich Reagent Plus grade). The diet was pelleted AIN93G semipurified diet (TD-94245) or pelleted commercial diet (TD-98186), both from Harlan Teklad (Madison, WI).

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All treatments were given for 7 days, after which the rats were fed tap water and the same diet as before, but without NaNO2 or hot dogs. Urine and, in one experiment, feces were collected daily on days 1–4. Day 1 was the 24 h period starting as soon as nitrite or hot dog feeding had stopped. Days 2–4 were the subsequent 24 h periods. For collecting the urine and feces, we used a rack of metabolic cages. Urine passed through wire mesh and a funnel into heavy test tubes, which were kept on ice until the urine was collected after 24 h. Feces was collected daily from the wire mesh. Urine and feces were collected on days 1–4, starting as soon as the treatments were stopped. Before each collection, the cages were washed with water but without soap. The urine was estimated to have been in contact with only 20% of the feces. The feces and diets appeared dry and were analyzed without first drying them. Urine and fecal samples were stored at −30 °C until they were analyzed.

2.2. Determination of ANC in Urine. ANC were determined using a thermal energy analysis (TEA) instrument (Advanced Chromatographic Systems, Charleston, SC).19,21 A mixture of 700 mL of urine, 100 mL of water, 100 mL of 0.1 N HCl and 100 mL of a freshly prepared saturated solution of sulfamic acid (SA) in water (SA reagent) was kept for up to 4 h on ice, and 100 mL samples of this mixture were injected into the TEA reaction vessel and analyzed for ANC.1,9 The TEA output was integrated with a model 202 chromatographic data system (Peak Simple, Torrance, CA) and expressed as nmol ANC/day. Standards with 0.1 nmol of N-nitrosopropoline were analyzed 3 times daily.

2.3. Determination of RSNO and RFeNO in Urinary ANC. This followed the method of Kuhnle et al.1,9 We first determined ANC in urine (Methods, Section 2.2). In other tubes, the 100 mL water added in the ANC method was replaced by 100 mL of 50 mM mercuric chloride in water for RSNO analysis or 100 mL of 50 mM potassium ferricyanide in water for RFeNO analysis. The resulting solutions were reacted for 30 min at room temperature, mixed with SA reagent and HCl, and analyzed for ANC. The results were subtracted from those for total ANC to give RSNO and RFeNO levels.

2.4. Determination of Urinary ANCP and Precursors of RSNO and RFeNO. To prepare nitrosated ANCP, a mixture of 200 mL of urine, 110 mL of 2 N HCl, 110 mL of 2 M NaNO2, and 1.58 mL of water was incubated for 1 h at 37 °C, mixed with 300 mL of SA reagent, kept for at least 5 min, diluted 10 or 100 times with tap water, and analyzed for ANC. To determine RSNO and RFeNO in the urinary ANCP, a solution of nitrosated urinated ANC was diluted 10 or 100 times with water, brought to pH 7 with sodium carbonate, and analyzed for total ANC, RSNO, and RFeNO.

2.5. Determination of ANC in Feces and Diet. Fecal and diet samples were not dried, unlike our previous practice, because ANC results were expressed as amount/day. Samples (500 mg) of feces or diet were soaked in 4 mL of water for 18 h at room temperature, vortexed four times for 30 s each, and centrifuged for 10 min at 9000g. Of the supernatant, 800 μL was mixed with 100 mL of 0.1 N HCl and 100 mL of SA reagent. After the mixtures were kept for 15 min at room temperature and up to 4 h on ice, 100 μL samples were analyzed for ANC by TEA.

2.6. Measuring ANC Formation in Rat Urine after Incubation with NaNO2. Urine was collected for 24 h from two rats maintained on semipurified diet. Rat 1 was untreated. Rat 2 was fed 1.0 g NaN02/L water for 7 days before its urine was collected. Urine samples were incubated with water or a NaNO2 solution (final nitrite level, 1.8 mM) for 7 days at 4 °C and were then analyzed for ANC.

2.7. Determination of Nitrite in Urine. Nitrite was determined by the Griess colorimetric method.21 One milliliter of rat urine, urine–water mixtures, or (in standards) a solution of 6.9 μg NaNO2/mL water was mixed with 2.5 mL of Griess reagent and kept for 20 min at room temperature. The resulting purple color was measured as Aλ405

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Table 1. Effect of Varied Doses of Sodium Nitrite on Urinary ANC Excretion on Days 1–4 after Nitrite Treatment Was Stopped

| NaNO2 (mg/L water) | diet     | no. of rats | day 1 (nmol/day) | day 2 (nmol/day) | day 3 (nmol/day) | day 4 (nmol/day) |
|--------------------|----------|-------------|------------------|------------------|------------------|------------------|
| 0                  | semipurified | 8           | 0.9 ± 0.3        | 2.0 (1.3, 2.6)   | 2.1 (1.7, 2.5)   | 2.0 (1.4, 2.5)   |
| 60                 | semipurified | 2           | 4.2 (3.6, 4.8)   | 2.4 (1.8, 3.0)   | 1.7 (1.4, 2.0)   | 0.9 (0.8, 1.0)   |
| 120                | semipurified | 2           | 5.8 (5.6, 6.1)   | 2.8 ± 0.3        | 2.7 ± 0.5        | 1.7 ± 0.5        |
| 240                | semipurified | 4           | 10 ± 3           | 5.2 (3.2, 7.2)   | 2.1 (1.4, 3.6)   | 0.8 (0.6, 0.9)   |
| 480                | semipurified | 2           | 30 (28, 32)      | 19 ± 5           | 6.3 ± 1.2        | 2.8 ± 1.4        |
| 1000               | semipurified | 4           | 37 ± 2           | 19 ± 5           | 6.3 ± 1.2        | 2.8 ± 1.4        |
| 0                  | commercial | 6           | 12 ± 4           | 15 (11, 19)      | 10 (8, 13)       | 9.2 (8, 9.6)     |
| 120                | commercial | 2           | 15 (11, 19)      | 10 (8, 13)       | 9.2 (8, 9.6)     | 8.7 (5.7, 11.7)  |
| 240                | commercial | 2           | 23 (16, 30)      | 13 (10, 16)      | 14 (14, 14)      | 11 (8, 15)       |

“In Tables 1, 2, and 5, results are shown as mean (individual values) for two results or as mean ± SD for more than two results.”

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40% hot dogs were fed together with 240 mg NaNO₂/L, the mean nanomoles of urinary ANC on day 1 was 25 nmol/day compared to 12 nmol/day for hot dogs alone (Table 2), 10 nmol/day for 240 mg NaNO₂ alone (Table 1), and 0.9 nmol/day for no treatment (Table 2). This indicates a simple additive effect for the two test materials with no indication that nitrite had produced ANC from ANCP in the hot dogs. Urinary ANC output on day 3/urinary ANC output on day 1 was 9% for 40% hot dogs fed with semipurified diet (Table 2), 22% for 240 mg NaNO₂/L fed with semipurified diet (Table 1), and 61% for 240 mg NaNO₂/L fed with commercial diet (Table 1). Hence, it appears that nitrite or the resulting ANC were cleared from the body more slowly when commercial diet was fed than when semipurified diet was fed. Feeding Johnsonville hot dogs yielded urinary ANC levels significantly different from those for Bar-S hot dogs (Table 2).

### 3.4. Statistical Analysis of Urinary ANC Results

Table 3 shows linear regression analyses (a) for the correlations between urinary ANC excretion measured as nmol ANC/day for rats eating semipurified or commercial diet with varied concentrations of NaNO₂ in drinking water and (b) for rats fed various proportions of hot dogs in the diet. Results are shown for days 1–4. PC SAS version 9.3 was used for all summaries and analyses.\(^{32}\) Parameter estimates for the intercept and regression coefficients are presented in addition to the \(R^2\) and \(p\) values.

The \(p\) values were <0.01 for the effect of varied NaNO₂ levels on all 4 days (days 1–4) when NaNO₂ was administered with semipurified diet, were significant only on day 1 when NaNO₂ was administered with commercial diet, and was <0.01 for days

### Table 2. Effect of Feeding Hot Dogs Mixed with Semipurified Diet (with or without NaNO₂ in Drinking Water) on Urinary Excretion of ANC 1–4 Days after Nitrite Treatment Was Stopped

| hot dogs | brand           | percent of diet | NaNO₂ (mg/L water) | no. of rats | urinary ANC (nmol/day) |
|----------|----------------|----------------|--------------------|------------|------------------------|
|          |                |                |                    |            | day 1 | day 2 | day 3 | day 4 |
|          |                |                |                    |            | 0.9 ± 0.3 | “” | 4.0 ± 1.6 | 2.3 (1.8, 2.8) | 1.8 (1.7, 1.9) | 1.0 (0.8, 1.1) |
|          | Bar-S          | 18             | 0                  | 4          | 12 (10, 14) | 4.2 (3.8, 4.5) | 1.1 (1.1, 1.2) |
|          | Bar-S          | 40             | 0                  | 2          | 13 (10, 17) | 4.8 (2.5, 7.2) | 1.5 (1.4, 1.5) | 1.9 (1.6, 2.1) |
|          | Bar-S          | 60             | 0                  | 2          | 25 ± 4 | 6.4 ± 1.7 |
|          | Johnsonville   | 18             | 0                  | 4          | 2.8 ± 1.4 | 0.9 ± 0.6 | 1.1 ± 0.7 |

Empty cells indicate that samples were not collected or were lost.

### Table 3. Significance of Correlations of Urinary ANC Output with NaNO₂ Dose or Percent Hot Dogs in the Diet 1–4 Days after Treatments Were Stopped

| material fed | diet   | day | regression coefficient | \(R^2\) | \(p\) value |
|--------------|--------|-----|------------------------|--------|------------|
| NaNO₂        | semipurified | 1   | 0.0375 | 0.918 | <0.01 |
|              |         | 2   | 0.0177 | 0.886 | <0.01 |
|              |         | 3   | 0.0050 | 0.853 | <0.01 |
|              |         | 4   | 0.0016 | 0.367 | 0.003 |
| NaNO₂        | commercial | 1   | 0.038 | 0.666 | <0.01 |
|              |         | 2   | 0.0073 | 0.116 | 0.28 |
|              |         | 3   | 0.004 | 0.055 | 0.46 |
|              |         | 4   | 0.003 | 0.025 | 0.62 |
| hot dogs     | semipurified | 1   | 0.226 | 0.870 | <0.01 |
|              |         | 2   | 0.0692 | 0.720 | <0.01 |
|              |         | 3   | 0.0081 | 0.161 | 0.123 |
|              |         | 4   | 0.0146 | 0.521 | 0.004 |

A\(_{320}\) – A\(_{420}\) was included to correct for turbidity. In a test with 6.9 \(\mu\)g of NaNO₂ in 1 mL of water, A\(_{320}\) was 0.47 and A\(_{420}\) was 0.02.

### RESULTS

#### 3.1. Subtracting Blank Values

All NOC, RSNO, and RFeNO results in the feeding experiments are listed without first subtracting blank values measured when NaNO₂ or hot dogs were administered. Blank values were generally less than 10% of those when NaNO₂ or hot dogs were fed.

#### 3.2. Urinary ANC Formation Due to Contamination of Urine with NaNO₂

We initially collected urine for ANC analysis from rats while they were fed NaNO₂ in their drinking water. However, we wondered whether drinking water containing NaNO₂ would spill from the feeding bottles into the urine and react there with urinary ANCP to form artifactual ANC. In fact, storage of 1.8 mM NaNO₂ in two rat urine samples for 7 days at 4°C led to the formation of 4.3 and 20.6 \(\mu\)M ANC. Hence, NaNO₂ can react with urinary ANCP to generate ANC. In subsequent experiments, urine was collected only after nitrite treatment was stopped.

#### 3.3. Urinary ANC in Rats Fed NaNO₂ and/or Hot Dogs

Table 1 shows urinary ANC output/day for 4 days (called day 1, day 2, etc.) after the NaNO₂ feeding was stopped. Urine collection was started immediately after NaNO₂ had been fed for 7 days. The rats were maintained on semipurified or commercial diet. Urine volume varied from 5 to 20 mL/day; the higher values are attributed to spilled drinking water (which, at this stage, did not contain nitrite) in the urine collection tubes. When increasing doses of NaNO₂ were administered to rats fed semipurified diet, urinary ANC excretion on day 1 rose from 0.9 nmol/day for untreated rats to a mean of 37 nmol/day for a dose of 1000 mg NaNO₂/L water. For the latter dose of NaNO₂ given with semipurified diet, ANC output on day 4 fell to 2.8 nmol (still 3.1 times the control value of 0.9 nmol/day; Table 1). Urinary ANC levels were elevated for all five nitrite doses on days 1 and 2, but only for 1000 mg NaNO₂/L on day 3.

When commercial diet was fed without nitrite, urinary ANC output was 12 nmol/day, 13 times the value for rats fed semipurified diet without nitrite (Table 1). This increased urinary ANC excretion may have been due to ANC in the commercial diet, which contained 0.93 nmol ANC/g diet compared to 0.14 nmol ANC/g semipurified diet. When 120 or 240 mg NaNO₂/L water was fed with commercial diet, the ANC results on day 1 were 2.3–2.6 times those for the same NaNO₂ dose given with semipurified diet.

When Bar-S hot dogs were fed to rats as mixtures of 18, 40, or 60% hot dogs in semipurified diet, the mean urinary ANC output increased on day 1 from 0.9 nmol/day in the absence of hot dogs to 13 nmol/day for 60% hot dogs (Table 2). When
1, 2, and 4 when varied proportions of hot dogs in the diet were fed.

3.5. Comparison between Urinary and Fecal ANC Outputs after Rats Were Fed NaNO₂. Urinary and fecal ANC excretions in the same rats and at the same time were compared 1–4 days after four rats were fed 1000 mg NaNO₂/L water for 7 days (Table 4). The 20 fecal collections on day 1 weighed 1.2 ± 0.5 g/sample (mean ± SD). On days 1 and 2 after nitrite treatment, urinary ANC output was about 60% of that of fecal ANC output.

Table 4. Comparison between Urinary and Fecal Excretions of ANC 1–4 Days after Four Rats That Were Maintained on Semipurified Diet Had Been Treated for 7 Days with 1000 mg NaNO₂/L Water

| test material | day | ANC in urine (nmol/day) | ANC in feces (nmol/day) | urinary ANC / fecal ANC |
|---------------|-----|------------------------|------------------------|------------------------|
| none          |     | 0.9 ± 0.3               | 0.6 ± 0.2              | 0.9 ± 0.4              |
| NaNO₂         | 1   | 37 ± 2                  | 115 ± 93               | 0.6 ± 0.4              |
| NaNO₂         | 2   | 19 ± 5                  | 49 ± 30                | 0.6 ± 0.5              |
| NaNO₂         | 3   | 6 ± 1                   | 29 ± 11                | 0.2 ± 0.1              |
| NaNO₂         | 4   | 2.8 ± 1.4               | 6 ± 3                  | 0.5 ± 0                |

*The urine results are also listed in Table 1. Mean values for urinary ANC/fecal ANC ratios were calculated from the individual ratios for each rat and are not the same as mean urinary ANC/mean fecal ANC for all rats.

3.6. RSNO and RFeNO as Possible Components of the Urinary ANC and ANCP. Because ANC can include RSNO and RFeNO, we determined RSNO and RFeNO in the ANC of urine samples (with 1 sample/rat) collected from 16 rats. The results were expressed as percentages of RSNO or RFeNO in the total ANC. The analyzed samples included (a) urine from rats that were fed 0, 240, or 480 mg NaNO₂/L water (10 samples) or fed 18, 40, or 60% hot dogs (6 samples), (b) urine that was analyzed for RSNO (10 samples) or RFeNO (6 samples), (c) urine from rats fed semipurified diet (12 samples) or commercial diet (4 samples), and (d) urine analyzed for ANC (12 samples) or ANCP (4 samples). Some urine samples were included in more than one category. The 16 urine samples showed RSNO + RFeNO levels that were 1 ± 4% (mean ± SD) of the total ANC in each sample. Hence, it appears that all of the urinary ANC were NOC and all the urinary ANCP were NOCP.

3.7. Urinary Excretion of ANCP. ANCP output on day 1 was measured in the urine of two rats/condition that were fed NaNO₂ and/or hot dogs or that were untreated (Table 5). The urine samples contained mean values of 11–18 (for semipurified diet) or 23–48 (for commercial diet) μmol ANCP/ day. For rats fed 240 mg NaNO₂/L water with semipurified diet, urinary ANCP excretion on day 1 was 18 μmol/day (Table 5), compared to 10 nmol/day for ANC (Table 1). Note that ANC are expressed as μmol/day, compared to nmol/day for ANC.

3.8. Search for Nitrite in Rat Urine. In view of the finding that NaNO₂ reacts with urinary ANCP to form ANC, we wondered whether some of the administered nitrite had been excreted in the urine and had reacted there to form the urinary ANC listed in Tables 1 and 4. This, rather than the urinary excretion of preformed ANC as assumed up to this point, could have been the origin of the urinary ANC found when NaNO₂ or hot dogs were fed (Tables 1, 2, and 4). To check this possibility, we used the Griess colorimetric method, based on the formation from nitrite of a red dye, to determine nitrite in the rat urine.

3.8.1. Stability of Nitrite on Storage of Urine Containing Nitrite. Nitrite in urine might have decomposed when the urine was stored at −30 °C so that nitrite might have been present only in freshly excreted urine. To test this view, 500 μL samples of three rat urine samples (pH 7.1–8.3) with added NaNO₂ (6.0 μg/mL urine) were kept for 7 days at 4, −20, or −80 °C. Analysis of these samples for nitrite by the Griess test showed A₄₂₀ − A₆₂₀ values of 0.30–0.32. Hence, nitrite in urine was stable even after storage for 7 days at 4 °C.

3.8.2. Search for Nitrite in 16 Urine Samples. Rats were treated for 7 days with 480 or 1000 mg NaNO₂/L drinking water (four rats/concentration). The rats were fed semipurified diet (10 rats) or commercial diet (6 rats). The urine samples were stored for 2–6 weeks at −30 °C and then analyzed for nitrite by the Griess reaction. The pH of six stored urine samples was 7.7 ± 0.5 (mean ± SD). The urine samples showed <0.5 μg NaNO₂/mL in 15 samples and 2 μg NaNO₂/mL in one sample. A₄₂₀ was 0.08–0.20. A₆₂₀ was 0.07–0.16. Hence, ingested NaNO₂ is not excreted unchanged in the urine of rats.

4. DISCUSSION

Feeding either NaNO₂ or hot dogs to rats produced large amounts of ANC in the urine for 1–4 days after the feeding stopped (Tables 1, 2, and 4). A total dose of 63 g NaNO₂/kg body weight was reported to not induce colon cancer in rats. Nevertheless, four lines of evidence suggest that NaNO₂ could induce colon cancer via the in vivo formation of carcinogetic NOC: (a) The consumption of nitrite-treated processed meat has been linked to the occurrence of colon cancer (Introduction, paragraph 3.8). (b) Ingested NaNO₂ increased the fecal excretion of ANC in mice, and the urinary and fecal excretion of ANC in rats (Tables 1 and 4). (c) ANC that was obtained by nitrosation of hot-dog-derived ANCP included colonic aberrant crypt foci (a putative precursor for colon cancer) in mice. (d) The Corpet group has convincing evidence linking hot dog-derived ANC with the induction of preneoplastic lesions in the rat colon (Introduction, paragraph 3).

A colorimetric assay for nitrite demonstrated that ingested NaNO₂ is not excreted unchanged in the urine of rats (Section 3.8.2). Our failure to detect nitrite in rat urine is consistent with a report that humans who ingested nitrate did not excrete nitrite in their urine, even though in humans (but not in rats) 5% of ingested nitrate is reduced to nitrite, mostly in the oral cavity.
When Bar-S hot dogs were fed to rats as 40–60% of a semipurified diet, mean urinary excretion of ANC was 12–13 nmol on day 1 and 4.2–4.8 nmol on day 2 (Table 2). Presumably, when hot dogs were fed, urinary ANC in excess of the background level of 0.9 nmol/day arose from ANC ingested in the hot dogs and, perhaps, by in vivo nitrosation of ANCP in the hot dogs. The presence of ANC in the urine suggests that, in the reported experiment, most tissues had been exposed to ANC, whereas the presence of ANC in the feces may signify only that the GI tract had been exposed to ANC. Accordingly, urinary ANC could be preferentially associated with cancer outside the GI tract.

Future epidemiological studies could investigate whether human urinary ANC excretion is correlated with (a) nitrate intake in drinking water and (b) cancer incidence. Studies on urinary ANC would be much easier to perform than those on fecal ANC. Because most NOC are carcinogens and ANC in urine appears to consist entirely of NOC, such exposure to processed meat could be a cause of cancers in organs (in addition to the colon) that have been linked with processed meat (Introduction, paragraph 2). Also, drinking water with high levels of nitrate in Iowa was linked to an increased incidence of colon cancer, provided that ascorbate intake was low and processed meat intake was high.

Mice fed NaN O showed significantly increased fecal ANC outputs down to a dose of 32 mg NaN O/L drinking water. In a human study, we found significantly increased urinary excretion of N-nitrosoproline after giving single doses of 100–400 mg nitrate as sodium nitrate. The observed in vivo nitrosation after human ingestion of nitrate probably occurred because 5% of ingested nitrate is reduced to nitrite, which nitrosated proline to form N-nitrosoproline.

We could not find previous reports on urinary ANC excretion in rodents. In the only report on urinary ANC excretion in humans, Lin et al. in 2002 assayed urinary ANC levels for male subjects from two areas in Southern China: one (the island of Nan’ao) with a high incidence of esophageal cancer and the other (Lufeng, on the mainland) with a low incidence of esophageal cancer. Mean dietary ANC intake by men was 4.2 ± 0.8 μmol/day in Nan’ao and 0.3 ± 0.1 μmol/day in Lufeng. Twelve-hour urine samples contained 0.04 ± 0.01 nmol ANC in Nan’ao and 0.02 ± 0.01 nmol ANC in Lufeng (mean ± SE). This was a significant difference with p < 0.01. The high ANC urinary output in Nan’ao was attributed to consumption of a diet containing relatively large concentrations of ANC.

Urinary excretion of ANC increased on days 1 and 2 when NaN O or hot dogs was fed and was about 60% of fecal ANC excretion when NaN O was fed (Tables 1, 2, and 4). This helps to explain how processed meat could be a cause of cancers in organs outside the GI tract, type 2 diabetes, and coronary heart disease (Introduction, paragraph 2). In conclusion, future epidemiological studies could survey ANC excretion in the urine of people showing elevated risks for colon cancer and other diseases that have been linked to the consumption of processed meat.

■ AUTHOR INFORMATION

Corresponding Authors
*E-mail: smirvish@unmc.edu.
*E-mail: mzahid@unmc.edu.

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■ ABBREVIATIONS

ANC, apparent N-nitroso compounds; ANCP, ANC precursors; GI, gastrointestinal; NOC, N-nitroso compounds; NOCP, NOC precursors; RFeNO, nitrosyl iron compounds; RSNO, nitrosothiols; SA, sulfamic acid; TEA, thermal energy analysis

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