Drug-likeliness approach of 2-aminopyrimidines as histamine H₃ receptor ligands

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Abstract: A small series of compounds containing derivatives of 2,4-diamino- and 2,4,6-triaminopyrimidine (compounds 2–7) was synthesized and tested for binding affinity to human histamine H₃ receptors (hH₃Rs) stably expressed in HEK-293 cells and human H₃Rs (hH₃Rs) co-expressed with Gq₁₁ and Gβγ₁₁ subunits in Sf9 cells. Working in part from the lead compound 6-(4-methylpiperazin-1-yl)-N²-(3-(piperidin-1-yl)propyl)pyrimidine-2,4-diamine (compound 1) with unsatisfactory affinity and selectivity to hH₃Rs, our structure-activity relationship studies revealed that replacement of 4-methylpiperazino by N-benzylamine and substitution of an amine group at the 2-position of the 2-aminopyrimidine core structure with 3-piperidinopropoxophenyl moiety as an hH₃R pharmacophore resulted in N⁴-benzyl-N²-(4-(3-piperidin-1-yl)propoxy)phenyl)pyrimidine-2,4-diamine (compound 5) with high hH₃R affinity (k₅ = 4.90 ± 1.25 nM) and H₃R receptor subtype selectivity of more than 6,500×. Moreover, initial metric analyses were conducted based on their target-oriented drug-likeliness for predictively quantifying lipophilicity, ligand efficiency, lipophilicity-dependent ligand efficiency, molecular-size-independent efficiency, and topological molecular polar surface. As to the development of potential H₃R ligands, results showed that integration of the hH₃R pharmacophore in hH₃R-affine structural scaffolds resulted in compounds with high hH₃R affinity (4.5–650 nM), moderate to low hH₃R affinity (4,500–30,000 nM), receptor subtype selectivity (ratio hH₃R/hH₃R; 8–6,500), and promising calculated drug-likeliness properties.

Keywords: histamine, H₃ receptors, H₃ receptors, drug-likeliness

Introduction

Histamine effects are known to be mediated through binding to four known histamine receptor subtypes designated as histamine receptors H₁ to H₄ (H₁R–H₄R), that belong to the family of G-protein-coupled receptors. H₃R and H₄R antagonists have been used for many years in clinical settings in the treatment of allergic conditions and gastric ulcers, respectively, while the potential clinical applications of human H₃R (hH₃R) and human H₄R (hH₄R) ligands are currently being extensively studied. H₄R, initially described in 1983, is mostly expressed in the brain.¹₂ H₃Rs are recognized as presynaptic autoreceptors that regulate the synthesis and release of histamine as well as heteroreceptors on non-histaminergic neurons modulating the release of many other important neurotransmitters such as acetylcholine, norepinephrine, dopamine and serotonin.³–⁵ Given their localization and ability to affect multiple neurotransmitter systems, it has been hypothesized that H₃R antagonist/inverse agonists could be suitable for the therapeutic management of central nervous system (CNS) disorders such as sleep disorders, Alzheimer’s disease, schizophrenia, narcolepsy, epilepsy, pain, and obesity.⁶–⁹
**H₃Rs are the most recently recognized receptor subtype belonging to the histamine receptor family. Despite their approximate 31% sequence homology (54% in transmembrane domains) and similarity in genomic structures, hH₃R and hH₄R differ significantly in expression patterns. Whereas hH₃R is mainly restricted to the brain, hH₄R expression is found mostly in the peripheral tissues and cells of the immune system.**

Individual expression patterns of hH₃R on various hematopoietic cells such as mast cells, basophils, eosinophils, T-cells, and dendritic cells suggests an essential role in the regulation of immune response and inflammation.⁸⁻¹⁰

Due to structural similarities between hH₃R and hH₄R, it is expected that several imidazole-containing H₃R ligands such as clobenpropit, imetit, and thioperamide also have a significant affinity for hH₄R.¹¹ 2,4-Diaminopyrimidines were first identified by chemists from Palau Pharma (Barcelona, Spain) and UCB Pharma (Brussels, Belgium), although without specific descriptions of pharmacological data. Pfizer (New York, NY, USA) scientists also published studies on related pyrimidines such as hH₃R ligands substituted with aliphatic heterocycles, while Johnson & Johnson (New Brunswick, NJ, USA) prepared carbon-analog 2-amino-4-alkyl-pyrimidines.¹³ Furthermore, current medicinal chemistry efforts are predominantly focused on selective acting G protein-coupled receptor ligands, particularly histamine hH₃ and hH₄ selective agonists/antagonists. Previous studies resulted in the development of 6-(4-methylpiperazin-1-yl)-N⁴(3-piperidin-1-yl)propyl)pyrimidine-2,4-diamine, which had unsatisfactory affinity and selectivity for hH₃R and hH₄R. The blueprint structure was constructed on a 2,4,6-triaminopyrimidine core containing a central 2,4,6-triaminopyrimidine as a conjugated donor/acceptor group; a basic group in the western part of the molecule and a lipophilic moiety in the eastern part (Figure 1).¹³⁻¹⁵⁻¹⁷

Based on these efforts in designing ligands for hH₃Rs and hH₄Rs, the main objective of the current study was to link a well-established hH₃R pharmacophore with an hH₄R core structure, namely 2-aminopyrimidine, to determine which structural elements are necessary to create affinity to hH₃Rs. As a result, a 2,4,6-triaminopyrimidine core element was chosen for further structural modifications. Affinity to hH₄R stably expressed in HEK-293 cells and in hH₃R transiently co-expressed with G₃₁₆ and G₄₁₂ subunits in Sf9 cells was measured according to previously established methods to explore the influence of modifications to basic and lipophilic moieties in the western and eastern parts of the molecule, respectively.¹⁸⁻¹⁹ Several approaches are described: modifying compound 1 and related structures involved in the exchange of the lipophilic basic residue at 4-position (eastern part) by N-benzylamine moieties lacking basic character; replacement of the basic 4-methylpiperazino side chain at 6-position (western part) by unbranched basic piperidines maintaining the comparable spacer length; and substitution of the amine group at 2-position of the 2-aminopyrimidine core structure (Figure 1). Moreover, initial metric analyses were conducted to predictively quantify lipophilicity (clogP), ligand efficiency (LE), lipophilicity-dependent ligand efficiency (LELP), lipophilic efficiency (LipE), and topological molecular polar surface (TPSA), important for verification of drug-likeness of novel compounds 2–7.

**Material and methods**

**General**

Educts, reactants, and solvents were commercially obtained from Merck (Darmstadt, Germany), Sigma-Aldrich (Sigma-Aldrich Chemie GmbH, Buchs, Switzerland), Alfa Aesar (Ward Hill, MA, USA), ABCR (Gainesville, FL, USA), and Acros Organics (Thermo Fisher Scientific, Waltham, MA, USA). The microwave oven used was a Biotage Initiator 2.0 (Biotage, Uppsala, Sweden). High-pressure amida was performed using a Haberautoklav Modell IV Typ 50S (Carl Roth® GMBH, Karlsruhe, Germany). Thin layer chromatography was performed with 0.2 mm thick silica gel foil 60 F₂₅₄ (Merck). For detection of precursors and final products, standard methods were performed: ultraviolet (UV) irradiation, ethanolic solution of ninhydrin reagent, and hot air. Mobile phases consisted of different mixtures of dichloromethane and methanol saturated with gaseous ammonia atmosphere. Preparative column chromatography was performed on silica gel with particle sizes of 63 μm–200 μm (Merck). Mixtures of dichloromethane and ammonia-saturated methanol were used as mobile phases.

**Figure 1** Rough blueprint for hH₃R ligands.
For flash chromatography, prepacked columns were used on an InteliFlash 310™ chromatography system (Varian Medical Systems Inc., Palo Alto, CA, USA). The adsorbent used was 50 μm silica gel (SuperFlash™, Varian). The used mobile phases were mixtures of dichloromethane and ammonia-saturated methanol or mixtures of methanol and triethylamine (99:1). Detection was carried out using an UV detector (wavelength: λ = 256 nm). Infrared spectroscopy was measured on a Bruker Alpha-T spectrometer (Bruker BioSpin MRI GmbH, Ettlingen, Germany). Intensities of the bands were characterized as follows: s for strong (very intensive), m for medium (medium intensity), and w for weak (low intensity). 1H NMR and 13C NMR spectra were recorded on a Bruker AM-250 (DPX-250) spectrometer (Bruker): (1H: 250.1 MHz; 13C: 62.9 MHz); a Bruker Avance 300 (1H: 300 MHz, 13C: 75.7 MHz); and a Bruker Avance 400 (1H: 400 MHz, 13C: 100.6 MHz). Perdeuterated and dried solvents (dimethyl sulfoxide [DMSO]-d6) were used.

1H NMR and 13C NMR data were reported in the following order. 1) Chemical shift (δ) in parts per million (ppm) downfield: chemical shift signals were calibrated to non-complete perdeuterated solvent as secondary signals (signals were calibrated to internal reference trimethylsilsilane). δ values were δ(1H) = 2.54 and δ(13C) = 40.45 in DMSO-d6 as solvent. 2) Signal multiplicity as follows: br = broad; s = singlet; d = doublet; dd = double doublet; pt = pseudo triplet; t = triplet; m = multiplet; approximate coupling constants (J) in hertz (Hz) and number and assignment of protons (prop = propane; pip = piperidine; Ar = areyl; ax = axial; eq = equatorial). Electrospray ionization mass spectrometry was performed on a Fisons VG Platform II mass spectrometer (Fisons Instruments Ltd, Manchester, UK) in positive polarity. Data were identified as mass number ([M+H]+) and relative intensity (%). Elemental analyses for C, H, and N were performed on a Foss Heraeus CHN-O-rapid elemental analyser (Foss Heraeus, Hanau, Germany). Melting points were determined on a Büchi® B-530 melting point apparatus (Buchi, Flawil, Switzerland).

Chemistry
Precursor syntheses of compounds P-I to P-XVII
The current study focused on the synthesis and pharmacological evaluation of piperidin-1-yl-propoxyphenyl- and 3-(4-methylpiperazin-1-yl)propoxyphenyl-substituted 2-aminopyrimidine derivatives. Consequently, the two basic linkers were introduced into the aromatic 2-aminopyrimidine skeleton while keeping the lipophilic element at 4-position with the 4-methylpiperazine or N2-benzylamine substituent constant. Furthermore, type and position of the side-chain moieties at the 2-amino position and 4- and 6-positions of the pyrimidine ring were systematically modified.

P-I: 3-(Piperidin-1-yl)propan-1-ol hydrochloride
Piperidine (20 g, 236 mmol), 3-chloropropene-1-ol (14.8 g, 157 mmol), K2CO3 (32.5 g, 235 mmol), and KI (26 g, 157 mmol) were refluxed in absolute acetone (500 mL) for six days. The mixture was cooled to room temperature, inorganic components were removed by filtration, and the filtrate was concentrated by distillation. Purification of the crude orange oil was achieved by distillation under reduced pressure (15 mbar, 90°C–105°C). The oily product was crystallized as a hydrochloride from 2-propanol in isopropanol 0.5N HCl (white solid, 16.5 g, 73% yield).

1H-NMR (DMSO-d6; 300 MHz): δ = 10.54 (br-s, 1 H, NH); 4.42 (s, br-s, 1 H, OH); 3.41 (t, J3/HH = 6.0 Hz, 2 H, prop-1H2); 3.37–3.33 (m, 2 H, pip-2,6Heq); 3.02–2.95 (m, 2 H, prop-3H2); 2.82–2.78 (m, 2 H, pip-2,6Hax); 1.88–1.64 (m; 7 H, prop-H2, pip-5, 3 H2, pip-4 Heq); 1.43–1.31 (m, 1 H, pip-4Hax) ppm. 13C-NMR (DMSO-d6; 300 MHz): δ = 57.98; 53.57; 51.95; 26.38; 22.27; 21.45 ppm. Electrospray ionization mass spectrometry (ESI MS): 143.8 (M + H) (100%).

P-II: 1-(3-chloropropyl)piperidine hydrochloride
3-(Piperidin-1-yl)propan-1-ol (P, 5 g, 27.8 mmol) was suspended in 100 mL absolute toluene and an excess of thionyl chloride (10 mL, 83 mmol) was added drop-wise under inert atmosphere at 0°C. Once the exothermic reaction decayed, the mixture was stirred for 3 h at 60°C. On completion of the reaction, thionyl chloride and toluene were distilled off. The crude product was re-crystallized from ethanol (beige solid, 5.24 g, 95% yield).

1H-NMR (DMSO-d6; 300 MHz): δ = 10.90 (br-s, 1 H, NH); 3.73 (t, J3/HH = 6.0 Hz, 2 H, prop-3H2); 3.40–3.35 (m, 2 H, pip-2,6Heq); 3.10–3.03 (m, 2 H, prop-1H2); 2.89–2.77 (m, 2 H, pip-2,6Hax); 1.87–1.65 (m; 5 H, pip-5, 3 H2, pip-4Heq); 1.41–1.31 (m, 1 H, pip-4Hax). 13C-NMR (DMSO-d6; 300 MHz): δ = 53.56; 51.94; 42.50; 26.13; 22.19 ppm. ESI MS: 161.6 (M + H) (100%).

P-III: 4-(3-(piperidin-1-yl)propoxy)benzonitrile
Precursor II (5 g, 25.2 mmol), 4-hydroxybenzonitrile (3.31 g, 27.8 mmol), K2CO3 (10.5 g, 76 mmol), and KI (4.2 g, 25.2 mmol) were refluxed in 100 mL absolute aceton for 48 hours. After cooling, inorganic compounds were removed by filtration and the filtrate was concentrated
to dryness. The residue was dissolved in dichloromethane, filtrated, and extracted with 2M NaOH solution. Organic phases were collected, washed with saturated NaCl solution, dried with MgSO₄ and concentrated to dryness (solid, 4 g, 61% yield).¹⁷

¹H-NMR (DMSO-d6; 250 MHz): d=7.73 (d, 3JHH =10 Hz, 2 H, Ar-2,6H); 7.13 (d, 3JHH =10 Hz, 2 H, Ar-3,5H); 4.07 (t, 3JHH =5 Hz, 2 H, prop-1H2); 2.38–2.30 (m, 6 H, pip-2,6H2, prop-3H2); 1.88–1.83 (m, 2 H, prop-2H2); 1.47–1.35 (m, 6 H, pip-3,4,5H2) ppm. ¹³C-NMR (DMSO-d6; 250 MHz): d=164.42; 134.17; 119.05; 115.53; 102.99; 22.61; 21.46 ppm. ESI MS: 245.0 (M + H) (100%).

P-IV: (4-(3-(piperidin-1-yl)propoxy)phenyl)methanamine
Benzonitrile (P-III, 4 g, 15.5 mmol) was suspended in 50 mL ammonia-saturated methanol in an autoclave with a catalytic amount of Raney nickel and hydrated at 5bar hydrogen pressure. The catalyst was removed by filtration with Celite® 535 coarse (Sigma) and the solvent was removed under reduced pressure. The product was used for the following reactions without any further purification (solid, 4 g, 98% yield).²⁰²³

¹H-NMR (DMSO-d6; 300 MHz): d=7.24 (d, 3JHH =9 Hz, 2 H, Ar-2,6H); 6.88 (d, 3JHH =9 Hz, 2 H, Ar-3,5H); 4.07 (t, 3JHH =6 Hz, 2 H, prop-1H2); 3.72 (s, 2 H, CH2NH2); 2.37–2.30 (m, 6 H, pip-2,6H2, prop-3H2); 1.86–1.74 (m, 2 H, prop-2H2); 1.51–1.43 (m, 4 H, pip-3,5H2); 1.37–1.35 (m, 2 H, pip-4H2) ppm. ¹³C-NMR (DMSO-d6; 300 MHz): d=160.30; 131.23; 128.07; 117.18; 69.02; 58.36; 57.24; 47.96; 29.47; 28.70; 27.26 ppm. ESI MS: 250.0 (M + H) (100%).

P-V: 6-chloro-N⁴-(4-(3-(piperidin-1-yl)propoxy)benzyl)pyrimidin-2,4-diamine
4,6-Dichloropyrimidine-2-amine (500 mg, 3 mmol) and precursor P-IV (757 mg, 3 mmol) were suspended in diisopropylamine (DIPA) (1 mL, 6.1 mmol) in a microwave vial. The reaction mixture was heated and stirred in a microwave oven at 130°C for 1 hour. Solvent was removed under reduced pressure. Purification was performed by multiple re-crystallizations in acetone (solid, 535 mmg, 73% yield).

¹H-NMR (DMSO-d6; 250 MHz): d=7.62 (s, 2 H, NH2); 7.23 (d, 3JHH =8 Hz, 2 H, Ar-2,6H); 6.87 (d, 3JHH =9 Hz, 2 H, Ar-3,5H); 6.43 (s, 1H, pyrimidine-H); 5.79 (s, 1 H, NH); 4.39 (s, 2H, methylene-H); 4.02–3.98 (m, 2 H, prop-1H2); 3.01–2.78 (m, 6 H, prop-3H2, pip-2,6H2); 2.03–1.70 (m, 2 H, prop-2H2); 1.66–1.47 (m, 4 H, pip-3,5H2); 1.28–1.25 (m, 2 H, pip-4H2) ppm. ESI MS: 376.5 (M + H) (100%).

P-VI: 3-(piperidin-1-yl)propanitrile
Piperidine (6 mL, 58.7 mmol), 3-chloropropanenitrile (4.6 mL, 59.7 mmol), K₂CO₃ (24 g, 176 mmol), and KI (10 g, 59 mmol) were refluxed in 50 mL absolute acetone for 48 hours. Inorganic compounds were filtrated and the solvent was evaporated under reduced pressure. Column chromatography was performed for further purification of the product (solid phase: silica gel; mobile phase: mixture of dichloromethane and ammonia-saturated methanol 9:1) (solid, 5.2 g, 64% yield).²⁰²³

¹H-NMR (DMSO-d6; 250 MHz): d=2.67–2.61 (m, 2 H, prop-2H2); 2.55–2.49 (m, 2 H, prop-3H2); 2.93–2.35 (m, 4 H, pip-2,6H2); 1.52–1.38 (m, 6 H, pip-3,4,5H2) ppm. ¹³C-NMR (DMSO-d6; 250 MHz): d=56.51; 54.10; 48.45; 30.01; 25.58; 24.15 ppm. ESI MS: 138.7 (M + H) (100%).

P-VII: 3-(piperidin-1-yl)propan-1-amine
Precursor P-VI (2 g, 14.5 mmol) was suspended in 50 mL ammonia-saturated methanol in an autoclave with a catalytic amount of Raney nickel and hydrated at 5bar hydrogen pressure for 24 hours. The catalyst was removed by filtration with Celite 535 coarse and the solvent was removed under reduced pressure. Column chromatography was performed for further purification of the product (solid phase: silica gel; mobile phase: mixture of dichloromethane and ammonia-saturated methanol 95:5: 9:1) (solid, 1 g, 49% yield).²⁰²³

¹H-NMR (DMSO-d6; 250 MHz): d=2.55–2.49 (m, 2 H, prop-3H2); 2.25–2.19 (m, 6 H, pip-2, 6H2 prop-1H2); 1.46–1.34 (m, 8 H, pip-3, 4, 5H2, prop-2H2) ppm. ¹³C-NMR (DMSO-d6; 250 MHz): d=56.50; 54.11; 48.46; 29.80; 25.58; 24.17 ppm. ESI MS: 142.9 (M + H) (100%).

P-VIII: 6-chloro-N⁴-(3-(piperidin-1-yl)propyl)pyrimidin-2,4-diamine
4,6-Dichloropyrimidine-2-amine (300 mg, 1.8 mmol) and precursor P-VII (260 mg, 1.8 mmol) were suspended in N,N-diisopropylethylamine (0.63 mL, 3.6 mmol) in a microwave vial. The reaction mixture was heated and stirred in a microwave oven at 130°C for 1.5 hours. Purification was performed by multiple recrystallizations in absolute acetone (solid, 315 mg, 37%).

¹H-NMR (DMSO-d6; 250 MHz): d=7.93 (br-s, 1 H, NH); 6.38 (br-s, 2 H, NH2); 5.75 (s, 1 H, Ar-H); 3.00–2.76 (m, 8 H, prop-1, 3H2, pip-2,6H2); 1.92–1.60 (m, 8 H, prop-2H2, pip-3,4,5H2) ppm. ¹³C-NMR (DMSO-d6; 250 MHz): d=164.01; 162.87; 122.88; 54.88; 53.88; 51.95; 23.36; 22.31; 21.41 ppm. ESI MS: 270.1 (M + H) (100%).
P-IX: 1-(3-(4-nitroxyenoxy)propyl)piperidine
Precursor P-II (7.4 g, 37 mmol), 4-nitrophenole (5.2 g, 37 mmol), K₂CO₃ (15.4 g, 111 mmol), and KI (6.2 g, 37 mmol) were refluxed in 100 mL absolute dimethylformamide for 72 hours. Inorganic compounds were removed by filtration, the solution was concentrated, and 100 mL water was added. The water phase was washed with the organic solvent ethyl acetate. The collected phases were washed with 50 mL 2N NaOH solution and then with 50 mL of saturated NaCl solution, dried with MgSO₄ and concentrated under reduced pressure (solid, 7.6 g, 77% yield).²⁴

1H-NMR (DMSO-d₆; 400 MHz): δ=8.15 (d, J₂/HH =9 Hz, 2 H, Ar-2,6H); 7.09 (d, J₂/HH =8 Hz, 2 H, Ar-3,5H); 4.11 (t, J₂/HH =6 Hz, 2 H, prop-1H2); 2.35 (t, J₂/HH =7 Hz, 2 H, prop-H2); 1.93–1.81 (m, 2 H, prop-2H2); 1.49–1.44 (m, 4 H, pip-3,5H2); 1.37–1.34 (m, 2 H, pip-4H2) ppm.

13C-NMR (DMSO-d₆; 250 MHz): δ=163.96; 140.61; 126.53; 114.79; 67.04; 59.67; 54.82; 25.99; 24.08; 20.60 ppm. ESI MS: 265.51 (M + H) (100%).

P-X: 4-(3-(piperidin-1-yl)propoxy)aniline
Precursor P-IX (7.5 g, 34 mmol) was suspended in 50 mL absolute ethanol and hydrated at 5bar hydrogen pressure for 2 h in an autoclave using a catalytic amount of Palladium on carbon (10%). The used catalyst was removed by filtration with Celite 535 coarse and the solvent was removed under reduced pressure (solid, 7.5 g, 94% yield).²⁰-²³

1H-NMR (DMSO-d₆; 250 MHz): δ=6.65 (d, J₂/HH =8 Hz, 2 H, Ar-2,6H); 6.53 (d, J₂/HH =8 Hz, 2 H, Ar-3,5H); 3.84 (t, J₂/HH =8 Hz, 2 H, prop-1H2); 2.38–2.32 (m, 6 H, prop-3H₂, pip-2, 6H₂); 1.84–1.76 (m, 2 H, prop-2H2); 1.51–1.37 (m, 6 H, pip-4H₂, pip-3,5H₂) ppm.

13C-NMR (DMSO-d₆; 250 MHz): δ=150.04; 142.21; 115.34; 55.32; 54.10; 26.54; 25.59; 24.14; 18.48 ppm. ESI MS: 235.2 (M + H) (100%).

P-XI: 6-chloro-N⁴-(4-(3-(piperidin-1-yl)propoxy)phenyl)pyrimidin-2,4-diamin
4,6-Dichloropyrimidine-2-amine (500 mg, 3 mmol) and precursor P-X (714 mg, 3 mmol) were suspended in DIPEA (2.1 mL, 12.2 mmol) in a microwave vial. The reaction mixture was heated and stirred in a microwave oven at 160°C for 3 hours. Solvent was removed under reduced pressure. Purification was performed by multiple recrystallizations in absolute acetone and ethanol (solid, 250 mg, 22% yield).

1H-NMR (DMSO-d₆; 400 MHz): δ=9.09 (s, 1 H, NH); 7.58 (s, 1 H, pyrimidine-H); 7.49 (d, J₂/HH =9 Hz, 2 H, Ar-2,6H); 6.85 (m, 2 H, Ar-3,5H); 4.36–4.32 (m, 2 H, prop-3H2); 3.45–3.33 (m, 6 H, prop-1H2, pip-2,6-H2); 1.49–1.17 (m, 2 H, prop-2H2); 1.07–1.01 (m, 6 H, pip-3,5H2) ppm.

13C-NMR (DMSO-d₆; 250 MHz): δ=162.86; 162.08; 157.88; 153.87; 133.18; 121.87; 114.65; 93.21; 65.39; 53.67; 52.89; 23.77; 22.81; 21.77 ppm. ESI MS: 361.8 (M + H) (100%).

P-XII: 3-(1-methylpiperazin-1-yl)pipra-1-ol
1-methylpiperazine (12.7 g, 127 mmol), 3-chloropropan-1-ol (85 g, 900 mmol), K₂CO₃ (18 g, 130 mmol), and KI (15 g, 90 mmol) were refluxed in absolute acetone for 6 d. Purification was performed with column chromatography (solid phase: silica gel; mobile phase: dichloromethane/ammonia-saturated methanol 9:1) and precipitated as salt of hydrochloric acid (solid, 13.7 g, 78% yield).¹⁷

1H-NMR (DMSO-d₆; 250 MHz): δ=4.43 (br-s, 1 H, OH); 3.40 (t, J₂/HH =9 Hz, 2 H, prop-1H2); 2.49–2.48 (m, 2 H, prop-3H2); 2.49–2.17 (m, 8 H, pipra-2,3,5,6H2); 2.12 (s, 3 H, CH3); 1.53 (qu, 2 H, prop-2H2) ppm. ESI MS: 159.22 (M + H) (100%).

P-XIII: Synthesis of 1-(3-chloropropyl)-1-methylpiperazine dihydrochloride
Thionyl chloride (5.6 mL, 77 mmol) was added dropwise to precursor P-XII (5 g, 25.7 mmol) and stirred at 60°C for 3 hours. No solvent was used. The excess of thionyl chloride was removed by distillation and the crude product was washed several times with toluene (solid, 4.4 g, 97% yield).¹⁷

1H-NMR (DMSO-d₆; 250 MHz): δ=3.76 (t, J₂/HH =6 Hz, 2 H, prop-1H2); 3.69–3.49 (m, 10 H, pipra-2,3,5,6H2); 2.83 (s, 3 H, CH3); 2.31–2.17 (m, 2 H, prop-2H2) ppm.

13C-NMR (DMSO-d₆; 250 MHz): δ=54.90; 48.66; 47.36; 42.15; 26.27 ppm. ESI MS: 177.33 (M + H) (100%).

P-XIV: 1-methyl-4-(3-(4-nitroxyenoxy)propyl)piperazine
Precursor P-XIII (1 g, 4.7 mmol), 4-nitrophenole (0.7 g, 4.7 mmol), K₂CO₃ (2 g, 14 mmol), and KI (0.8 g, 4.7 mmol) were refluxed in 30 mL absolute dimethylformamide for 72 hours. Inorganic impurities were removed by filtration, the solvent was evaporated under reduced pressure, and 100 mL water was added. The water phase was extracted with dichloromethane, then collected organic phases were washed with 2N NaOH solution and saturated NaCl, dried with Na₂SO₄, and concentrated to dryness (solid, 600 mg, 46% yield).²⁴

1H-NMR (DMSO-d₆; 400 MHz): δ=7.95 (d, J₂/HH =8 Hz, 2 H, Ar-2,6H); 6.88 (d, J₂/HH =9 Hz, 2 H, Ar-3,5H); 3.91–3.88 (m, 2 H, prop-1H2); 2.29–2.27 (m, 2 H, prop-3H2); 2.18–1.09 (m, 8 H, pipra-H); 1.90 (s, 3 H, CH3); 1.69–1.62
P–XV: 4-(3-(1-methylpiperazin-1-yl)propoxy)aniline
Precursor P-XIV (7.5 g, 34 mmol) was suspended in 50 mL of absolute ethanol in an autoclave. A catalytic amount of Pd/C (10%) was added and hydration was performed under 5 bar hydrogen pressure for 24 hours. The catalyst was removed by filtration with Celite 535 coarse and the solvent was evaporated under reduced pressure (solid, 490 mg, 98%).

1H-NMR (DMSO-d6; 250 MHz): δ=6.63 (d, 3JHH =9 Hz, 2 H, Ar-2,6H); 6.50 (d, 3JHH =8 Hz, 2 H, Ar-3,5H); 4.58 (s, 2 H, NH2); 3.84 (t, 3JHH =6 Hz, 2 H, prop-1H2); 2.41–2.32 (m, 10 H, piperazine-H, prop-3H2); 2.14 (s, 3 H, CH3); 1.88–1.73 (m, 2 H, prop-2H2) ppm. 13C-NMR (DMSO-d6; 250 MHz): δ=149.95; 142.38; 115.38; 66.35; 54.75; 54.53; 45.72; 26.44 ppm. ESI MS: 250.0 (M + H) (75%).

P–XVI: N-benzyl-2-chloropyrimidin-4-amine
2,4-Dichloropyrimidine (1 g, 6.7 mmol) and benzylamine (1.3 g, 10 mmol) were dissolved (0.7 mL, 6.7 mmol) and refluxed in isopropanol. Solvent was evaporated and the resulting residue was dissolved in dichloromethane, washed with saturated NaHCO3 and NaCl solution, dried with MgSO4, and concentrated to dryness. The crude product was purified with flash chromatography (solid phase: silica gel; mobile phase: dichloromethane/methanol (1% triethylamine, 97:3) (solid, 975 mg, 66% yield).

1H-NMR (DMSO-d6; 250 MHz): δ=8.44 (s, 1 H, NH); 8.00 (d, 3JHH =6 Hz, 1 H, pyrimidine-6H); 7.41–7.33 (m, 5 H, Ar-H); 6.58 (d, 3JHH =5 Hz, 1 H, pyrimidine-H); 4.55 (s, 2 H, methylene-H2) ppm. 13C-NMR (DMSO-d6; 300 MHz): δ=134.56; 129.44; 129.12; 128.38; 127.71; 127.63; 127.35; 53.18 ppm. ESI MS: 220.19 (M + H) (100%).

P–XVII: 2-chloro-4-(4-methylpiperazin-1-yl)pyrimidine
2,4-Dichloropyrimidine (500 mg, 3.4 mmol), 1-methylpiperazine (340 mg, 3 mmol), and DIPEA (0.9 mL, 6.7 mmol) were dissolved in absolute ethanol in a microwave vial. The reaction mixture was heated and stirred at 70°C for 45 min. The solvent was removed under reduced pressure. The resulting residue was dissolved in 20 mL dichloromethane, washed with 20 mL water, dried with Na2SO4, and finally concentrated to dryness (solid, 347 mg, 18% yield).

1H-NMR (DMSO-d6; 250 MHz): δ=8.05 (d, 3JHH =6 Hz, 1 H, pyrimidine-5H); 6.82 (d, 3JHH =6 Hz, 1 H, pyrimidine-6H); 3.62–3.59 (m, 4 H, piperazine-2,6H2); 2.41–2.36 (m, 4 H, piperazine-3,5H2); 2.21 (s, 3 H, CH3 ppm. 13C-NMR (DMSO-d6; 250 MHz): δ=171.02; 167.79; 98.83; 54.77; 48.49 ppm. ESI MS: 413.16 (M + H) (100%).

Synthesis of final products (compounds 2–7)

Compound 2: 6-(4-methylpiperazin-1-yl)-N⁴-(3-(piperidin-1-yl)propyl)pyrimidin-2,4-diamine
Precursor P-VIII (200 mg, 0.7 mmol) and 1-methylpiperazine (0.17 mL, 1.5 mmol) were suspended in DIPEA (0.25 mL, 1.5 mmol) in a microwave vial. The reaction mixture was heated and stirred at 160°C for 3 hours. Purification was performed by flash chromatography (solid phase: silica gel; mobile phase: dichloromethane/ammonia-saturated methanol 97:3, 9:1). For elementary analysis and melting point determination, the product was precipitated as a salt of maleic acid (solid, 140 mg, 56% yield).

IR (KBr): δ=3,390.47; 2,944.17 m; 2,697.79 m; 1,574.12 s; 1,384.70 s; 1,259.92 m; 11,94.27 m; 864.25 m; 711.22 m; 369.50 w cm⁻¹. 1H-NMR (DMSO-d6; 400 MHz): δ=6.15 (s, 1 H, NH); 5.46 (s, 2 H, NH2); 5.01 (s, 1 H, Ar-H); 3.36 (t, 4 H, piperazine-2,6H2); 3.17–3.12 (m, 2 H, prop-1H2); 2.32–2.25 (m, 10 H, pip-2,6H2, prop-3H2, piperazine-3,5H2); 1.62–1.59 (m, 2 H, prop-2H2); 1.52–1.47 (m, 4 H, pip-3,5H2); 1.39–1.38 (m, 2 H, pip-4H2) ppm. 13C-NMR (DMSO-d6; 250 MHz): δ=167.11; 134.84; 103.49; 56.70; 55.12; 53.74; 52.18; 52.08; 24.66; 23.33; 21.19 ppm. ESI MS: 334.2 (M + H) (100%) base. Analytical calculation (Anal Calc) for C17H17ClN2×3C4H9O×2H2O: C, 49.68; H, 5.94; N, 12.29. Found: C, 49.94; H, 6.15; N, 13.98.

Compound 3: 6-(4-methylpiperazin-1-yl)-N⁴-(4-(3-(piperidin-1-yl)propoxy)phenyl)pyrimidin-2,4-diamine
Precursor P-XI (235 mg, 0.65 mmol) and 1-methylpiperazine (0.15 mL, 1.3 mmol) were suspended in DIPEA (0.22 mL, 1.3 mmol) in a microwave vial. The reaction mixture was heated and stirred at 160°C for 3 hours. The solvent was removed under reduced pressure and the product was precipitated multiple times as a hydrochloride salt in absolute ethanol (solid, 91 mg, 33% yield).

IR (KBr): δ=3,283.09 m; 2,935.32 s; 2,522.62 s; 1,573.20 s; 1,505.95 s; 1,291.06 m; 1,218.53 s; 1,015.95 s; 1,002.40 m; 837.23 s; 797.83 s cm⁻¹. 1H-NMR (DMSO-d6; 250 MHz): δ=8.67 (s, 1 H, NH); 7.54 (d, 3JHH =9 Hz, 2 H, Ar-2,6H); 6.85 (d, 3JHH =9 Hz, 2 H, Ar-3,5H); 5.82 (br-s, 2 H, NH2); 5.35 (s, 1 H, pyrimidine-H); 4.44 (s, 4 H, piperazine-3,5H2); 4.04 (t, 3JHH =6 Hz, 2 H, prop-3H2); 3.23 (s, 4 H, piperazine-2,6H2); 3.04 (m, 6 H, prop-1H2, pip-2,6H2); 2.35 (s, 3 H, CH3).
Compound 4: 6-(4-methylpiperazin-1-yl)-N-(4-(3-
(piperidin-1-yl)propoxy)benzyl)pyrimidin-2,4-diamine Precursor P-V (750 mg, 2 mmol) and 1-methylpiperazine (0.29 mL, 2.6 mmol) were suspended in DIPEA (0.68 mL, 4 mmol) in a microwave vial. The reaction mixture was heated and stirred at 160°C for 2 hours. Purification was performed by flash chromatography (solid phase: silica gel; mobile phase: dichloromethane/methanol with 1% triethylamine; 99:1, 85:15). The crude product was dissolved in dichloromethane, washed with 2M NaOH solution, dried with MgSO4, and concentrated to dryness under reduced pressure (solid, 117 mg, 13% yield).

IR (KBr): ν = 3577.41 m; 2694.43; 1583.63 s; 1385.07 s; 1257.6 m; 1193.36 m; 973.95 s; 864.27 m; 710.27 m; 558.81 m cm⁻¹. ¹H-NMR (DMSO-d6; 400 MHz): δ = 7.20 (d, 3JHH = 9 Hz, 2 H, Ar-2,6H); 6.85 (d, 3JHH = 9 Hz, 2 H, Ar-3,5H); 5.30 (s, 1 H, NH); 5.07 (s, 1 H, pyrimidine-H); 4.32 (s, 2 H, methylene-H2); 3.96 (t, 3JHH = 6 Hz, 2 H, prop-3H2); 3.35 (pt, 3JHH = 5 Hz, 6 H, piperazine-2,6H2); 2.38–2.29 (m, 10 H, prop-1H2, pip-2,6H2, pip-3H2); 2.19 (s, 3 H, CH3); 1.86–1.82 (m, 2 H, prop-2H2); 1.52–1.38 (m, 6 H, pip-3,4,5H2) ppm. ¹³C-NMR (DMSO-d6; 250 MHz): δ = 164.28; 162.45; 162.35; 157.33; 132.48; 128.29; 114.67; 65.86; 55.15; 54.42; 54.07; 45.48; 43.55; 26.28; 25.55; 24.10 ppm. ESI MS: 440.5 (M + H) (100%). Maldi HRMS: 440.3127 (M + H) (100%); calc. 440.3132 (M + H) (100%).

Compound 5: N⁴-benzyl-N²-(4-(3-(piperidin-1-yl)
proproxy)phenyl)pyrimidin-2,4-diamine The precursors P-X (340 mg, 1.5 mmol) and P-XVI (435 mg, 1.9 mmol) were suspended in trifluoroacetic acid (0.36 mL, 4.6 mmol) in a microwave vial under inert gas atmosphere. The reaction mixture was heated and stirred at 140°C for 1 hour. The solvent was removed under reduced pressure. The crude product was purified by flash chromatography (solid phase: silica gel; mobile phase: dichloromethane/methanol with 1% triethylamine; 9:1) and recrystallized multiple times in absolute ethanol. For elementary analysis and melting point determination, the product was precipitated as salt of maleic acid (solid, 92 mg, 14% yield).

IR (KBr): ν = 2950.06 w; 1508.42 s; 1356.11 s; 1238.79 m; 1053.26 m; 864.63 m; 827.81 s; 731.88 s; 568.83 s cm⁻¹. ¹H-NMR (DMSO-d6; 400 MHz): δ = 8.80 (s, 1 H, NH); 7.87 (d, 3JHH = 8 Hz, 1 H, pyrimidine-6H); 7.67–7.63 (m, 2 H, Ar-2,3,5H); 7.34–7.23 (m, 5 H, Ar1-H); 6.583 (d, 3JHH = 9 Hz, 2 H, Ar-2,6H2); 6.04–5.99 (m, 1 H, pyrimidine-5H); 5.84 (s, 2 H, methylene-H2); 4.03–3.99 (m, 2 H, prop-1H2); 2.46–2.36 (m, 6 H, pip-2,6H2, prop-3H2); 1.93–1.89 (m, 2 H, prop-2H2); 1.60–1.46 (m, 6 H, pip-3,4,5H2) ppm. ¹³C-NMR (DMSO-d6; 250 MHz): δ = 164.44; 158.57; 158.16; 157.74; 139.14; 129.43; 129.11; 128.26; 127.18; 115.46; 114.46; 65.11; 53.44; 48.53; 41.23; 22.43; 10.89; 8.43 ppm. ESI MS: 418.5 (M + H) (100%). Maldi HRMS: 433.2708 (M + H) (100%); calc. 433.2710 (M + H) (100%).

Compound 7: 4-(4-methylpiperazin-1-yl)-N-(4-(3-
(methylpiperazin-1-yl)propoxy)phenyl)pyrimidin-2-amine Precursors P-XV (150 mg, 0.7 mmol) and P-XVII (176 mg, 0.7 mmol) were suspended in trifluoroacetic acid (0.16 mL, 2.1 mmol) in a microwave vial under inert gas atmosphere. The reaction mixture was heated and stirred at 140°C for...
1 hour. The solvent was removed under reduced pressure. The crude product was purified by flash chromatography (solid phase: silica gel; mobile phase: DCM/MeOH with 1% triethylamine; 9:1) and recrystallized multiple times in absolute ethanol. For determination of elementary analysis and melting point, the final product was precipitated as a salt of maleic acid (solid, 174 mg, 59% yield).

\[ 1^1H-NMR \text{ (DMSO-d6; 250 MHz): } d=8.85 \text{ (br-s, 1 H, NH);} \]
\[ 7.93 \text{ (d, J_HH=6 Hz, 1 H, pyrimidine-5H), 7.58 \text{ (d, J_HH=5 Hz, 2 H, Ar-3,5H);} \]
\[ 6.83 \text{ (d, J_HH=9 Hz, 2 H, Ar-4,6H);} \]
\[ 6.22 \text{ (d, J_HH=6 Hz, 1 H, pyrimidine-6H);} \]
\[ 3.95 \text{ (t, J_HH=6 Hz, 2 H, prop-1H2);} \]
\[ 3.72–3.61 \text{ (m, 6 H, prop-3H2, piperazine-2,3,5H2);} \]
\[ 2.68–42.62 \text{ (m, 10 H, piperazine-1,2,6H2, piperazine-2,3,5,6H2);} \]
\[ 2.27 \text{ (s, 3 H, CH3);} \]
\[ 1.93–1.83 \text{ (m, 2 H, prop-2H2) ppm.} \]

ESI MS: 426.7 (M + H) (100%). Maldi HRMS: 426.297 (M + H) (100%); calc. 426.2976 (M + H) (100%).

**Pharmacology**

**hH4R binding assay**

Radioligand binding assays were performed on HEK-293 cell membranes as previously described with slight modifications. Shortly before beginning the experiments, cell membrane preparations were thawed, homogenized by sonification at 4°C, and kept in ice-cold binding buffer (12.5 mM MgCl2, 100 mM NaCl, 75 mM Tris/HCl; pH 7.4). Competitive binding experiments were performed as follows: membranes (20–25 µg/well in a final volume of 0.2 mL binding buffer) were incubated with [3H]Histamine (10 nM, 85 Ci/mmol) and different concentrations of test ligands. Assays were run at least in duplicates with seven concentrations of 0.01 nM–100 µM of the test compound. Incubations were performed for 90 min at 25°C and shaking at 250 rpm. Non-specific binding was determined in the presence of 100 µM unlabeled histamine. Bound radioligand was separated from free radioligand by filtration through GF/B filters prepared from Sf9 cells expressing hH4Rs and co-expressed with G protein Gα12 and Gβγ12 subunits in a final volume of 0.2 mL containing binding buffer and [3H]Histamine (10 nM, 15.3 Ci/mmol). Assays were run in triplicate with seven concentrations of 0.1 nM–100 µM of the test compound. Incubations were performed for 60 min at 25°C and shaking at 250 rpm. Non-specific binding was determined in the presence of 100 µM unlabeled histamine. Bound radioligand was separated from free radioligand by filtration through GF/B filters pretreated with 0.3% (m/v) polyethyleneimine and washed three times with 5 mL of ice-cold binding buffer (4°C). The amount of radioactivity collected on the filter was determined by liquid scintillation counting. Competitive binding data were analyzed by GraphPad Prism 3.02 (GraphPad) software using non-linear least squares fit. Ki values were calculated from the IC50 values according to the Cheng–Prusoff equation.

**hH4R binding assay**

Radioligand binding assays were performed on Sf9 cell membranes as previously described with slight modifications. Prior to the experiments, cell membranes were sedimented by a 10 minutes centrifugation at 4°C and 16,000 g and resuspended in binding buffer (12.5 mM MgCl2, 1 mM EDTA, 75 mM Tris/HCl, pH 7.4). Competitive binding experiments were conducted in incubating membranes with 35 µg/well prepared from Sf9 cells expressing hH4R and co-expressed with G protein Gα12 and Gβγ12 subunits in a final volume of 0.2 mL containing binding buffer and [3H]Histamine (10 nM, 15.3 Ci/mmol). Assays were run in triplicate with seven concentrations of 0.1 nM–100 µM of the test compound. Incubations were performed for 60 min at 25°C and shaking at 250 rpm. Non-specific binding was determined in the presence of 100 µM unlabeled histamine. Bound radioligand was separated from free radioligand by filtration through GF/B filters pretreated with 0.3% (m/v) polyethyleneimine and washed three times with 5 mL of ice-cold binding buffer (4°C). The amount of radioactivity collected on the filter was determined by liquid scintillation counting. Competitive binding data were analyzed by GraphPad Prism 3.02 (GraphPad) software using non-linear least squares fit. Ki values were calculated from the IC50 values according to the Cheng–Prusoff equation.

**Metric analyses**

The values for LE, LELP, and LipE were calculated for compounds 1–7 based on their affinities for hH4Rs and hH4Rs to further corroborate their drug-like nature and assessment as potential new leads (Table 2). These widely-used three metrics are important models in drug discovery and are easy to use for assessing whether a ligand derives its potency from optimal fit with a target protein or simply by making many contacts. Values for LE, LELP, and LipE were calculated using equations 1, 2, and 3, respectively:

\[ \text{LE} = -0.592 \cdot \ln(K_i)/n^2 \text{ of heavy (non-hydrogen) atoms} \]

\[ \text{LELP} = \text{clogP/LE} \]

\[ \text{LipE} = \text{pK}_i - \text{clogP} \]

\[ \text{clogP} = \ln(C_{50})/n^2 \text{ of heavy (non-hydrogen) atoms} \]
An additional molecular descriptor, polar surface area (PSA), has been shown to correlate very well with human intestinal absorption.\textsuperscript{33} PSA is very suitable for predicting drug transport properties.\textsuperscript{33} In the current study, topological PSA (TPSA) was calculated using Molinspiration Depiction Software (Molinspiration Cheminformatics, Slovensky Grob, Slovak Republic).\textsuperscript{34} The procedure was based on the summation of the tabulated surface contributions of polar fragments, ie atoms regarding their physicochemical environment. Fragment contributions were determined by least squares fitting to the single conformer 3D PSA for 34,810 drugs from the World Drug Index. TPSA data usually provides results of virtually the same quality as classical 3D PSA; calculations, however, were 2–3 orders of magnitude faster.\textsuperscript{35,36}

**Results**

**Chemistry**

**Synthesis of precursors PI–XVII**

In the present study, the precursors PI–IV, PVI–VII, and PX–XI containing piperidine were prepared according to previously described methods.\textsuperscript{9,20–23} The precursors PV, PVIII, and PXXI were achieved through the nucleophilic reaction of 4,6-dichloropyrimidin-2-amine with precursors PIV, PVII, and PX, respectively, applying previously developed synthetic methodologies.\textsuperscript{17} The 4-methylpiperazine substituted precursors PXII–PXV and precursors PXVI and PXVII bearing benzyl or pipеразин-1-yl at 4-position amine moiety were prepared in good yields.\textsuperscript{22}

**Synthesis of final compounds 2–7**

Compounds 2–7 were prepared in moderate to good yields using well-established conditions for microwave assisted nucleophilic substitution reactions of 4,6-dichloropyrimidin-2-amine and an appropriate basic side chain. As polar, protic solvent 2-propanol or ethanol was used and DIPEA acted as a basic auxiliary. Accordingly, precursors PVIII, PXI, and PV were intermediates for the final compounds 2, 3, and 4, respectively, while PXV was the starter precursor for final compounds 6 and 7. Finally, the nucleophilic reaction of PX with PXVI yielded compound 5. All resulting precursors PI–XVII and final compounds 2–7 described in the current study were synthesized by a series of reactions as shown in Figures 2–4.

**Pharmacology**

**H\textsubscript{3}R and H\textsubscript{2}R binding affinity**

The final compounds 2–4 and 7 resulted from the combination of hH\textsubscript{3}R pharmacophore with 4-(4-methylpiperazin-1-yl)pyrimidin-2-amine. hH\textsubscript{2}R affinities of compounds 2–7 were evaluated by competition binding experiments with [\textsuperscript{3}H]N\textsubscript{\textalpha}{4\texttextsubscript{methylhistamine (MeHA) on membranes of HEK-293 cells stably expressing hH\textsubscript{2}R. Compounds 2–7 were further tested for their affinity to hH\textsubscript{2}R by a [\textsuperscript{3}H] histamine replacement assay on membranes of Sf9 cells transiently co-expressing hH\textsubscript{2}R with G\textsubscript{ass} and G\textsubscript{12} subunits.\textsuperscript{9,18–20} The H\textsubscript{2}R affinity observed for compounds 2–7 was in the nanomolar concentration range (K\textsubscript{i} values of 4.49–649 nM), whereas affinity for hH\textsubscript{2}R was found to be evidently lower (K\textsubscript{i} values of 4,480–49,240 nM) (Table 1).

**Metric analyses**

Values for LE (hH\textsubscript{3}R), LELP (hH\textsubscript{3}R), and LipE (hH\textsubscript{3}R) for compounds 1–7 were found to be in the range of 0.27–0.37, 1.39–13.55, and 2.69–5.34, respectively. Also, values for compounds 1–7 were in the range of 0.21–0.25, 2.04–17.82, and 0.20–3.49 for LE (hH\textsubscript{3}R), LELP (hH\textsubscript{3}R), and LipE (hH\textsubscript{3}R), respectively. Furthermore, the values describing the topological molecular polar surface were in the range of 60.00 Å\textsuperscript{2}–82.78 Å\textsuperscript{2}. Results of pharmacological screening and calculated physicochemical parameters are summarized in Tables 1 and 2.

**Discussion**

The synthesis of compound 2 should provide insight into whether the 2-aminopyrimidin ring can replace the aromatic center of the hH\textsubscript{3}R pharmacophore, namely piperidin-1-yl-propanyloxyphenyl. Indeed, compound 2 containing piperidinopropyl as the H\textsubscript{3}R pharmacophore observed for hH\textsubscript{2}R (K\textsubscript{i} (hH\textsubscript{2}R) = 49,240 nM) was decreased by approximately 2.3 × compared to the lead structure 1 (K\textsubscript{i} (hH\textsubscript{2}R) = 118,295 nM), despite keeping the entire 4-(4-methylpiperazin-1-yl)pyrimidin-2-amine moiety at 4-position in the eastern part of its structure. However, affinity for hH\textsubscript{2}R was only in the three-digit nanomolar range (K\textsubscript{i} (hH\textsubscript{2}R) = 580 nM) and was thus not satisfactory. Moreover, previous structure-activity relationships (SARs) of H\textsubscript{3}R antagonists/inverse agonists have shown that the presence of an ether function in the hH\textsubscript{3}R pharmacophore is necessary for producing high affinity for hH\textsubscript{3}R.\textsuperscript{8,11} The lack of an ether function, which was replaced by an amine moiety in compound 2, could explain the observed low affinity for hH\textsubscript{2}R (K\textsubscript{i} [hH\textsubscript{2}R] = 580 nM). As a result, inserting an aromatic ether linkage connecting the basic piperidinylpropyl moiety with the 4-(4-methylpiperazin-1-yl)pyrimidin-2-amine structural core resulted in compound 3 with a significant 49× increased affinity for hH\textsubscript{2}R (K\textsubscript{i} [hH\textsubscript{2}R] = 11.83 nM) and an hH\textsubscript{2}R/hH\textsubscript{3}R ratio of 1,232 towards the H\textsubscript{3}R subtype.
Figure 2 Synthesis of precursors PI–PXi.
Notes: Reagents and conditions. 1a) K₂CO₃, KI, acetone, reflux, 6 days; 1b) HCl, 2-propanol; 79%; 1c) 4-hydroxybenzonitrile, K₂CO₃, KI, acetone, reflux, 2 days; 1d) Raney-nickel, ammonia-saturated methanol, autoclave, 5bar hydrogen pressure; 1e) 4,6-Dichloropyrimidin-2-amine, DIPEA, isopropanol, MW, 130°C, 1 hour. 2a) 3-Chloropropannitrile, K₂CO₃, KI, acetone, 2 days. 2b) Raney-nickel, ammonia-saturated methanol, autoclave, 5bar hydrogen pressure. 2c) 4,6-dichloropyrimidin-2-amine, DIPEA, isopropanol, MW, 130°C, 1 hours. 3a) 4-Nitrophenole, K₂CO₃, KI, dimethylformamide, reflux, 3 days. 3b) Pd/C (10%), ethanol, autoclave, 5bar hydrogen pressure, 2 days. 3c) 4,6-Dichloropyrimidin-2-amine, DIPEA, isopropanol, MW, 160°C, 3 hours.

Figure 3 Synthesis of precursors PXII–PXVII.
Notes: Reagents and conditions. 1a) 3-Chloropropan-1-ol, K₂CO₃, KI, acetone, reflux, 6 days. 1b) SOCl₂, 60°C, 3 hours. 1c) 4-Nitrophenole, K₂CO₃, KI, dimethylformamide, reflux, 3 days. 1d) Pd/C (10%), ethanol, autoclave, 5bar hydrogen pressure, 2 days. 2a) 2,4-Dichloropyrimidine, isopropanol, reflux, 1 hour for PXVI. 2b) 2,4-Dichloropyrimidine, DIPEA, ethanol, MW, 70°C, 45 minutes for PXVII.
Conversely, the side chain elongation in the western part of the structure performed with the design and synthesis of compound 4 showed a nearly 5× weaker affinity for hH3Rs (Ki [hH3R] = 49.11 nM) and a 3× higher affinity for hH3Rs (Ki [hH3R] = 4,480.83 nM) compared to compound 3, indicating that such a substitution was unsuccessful and has not been explored any further. Moreover, previous studies of the SAR of H3R antagonists revealed that an amine moiety at 2-position in the 2-aminopyrimidine H3R pharmacophore was necessary to elicit affinity for hH3R.24 Thus, the substitution of the amine group at 2-position with the piperidinylpropyloxy H3R pharmacophore together with replacement of the 4-methylpiperazin-1-yl moiety in the eastern part by N-benzylamine significantly

Figure 4 Synthesis of final compounds 2–7.
Notes: Reagents and conditions. 1) 1-Methylpiperazine, DIPEA, MW, 160°C, 3 hours for 2 and 3, 2 hours for 4. 2) Precursor PXVI, trifluoroacetic acid, MW, 140°C, 1 hour. 3a) Precursor PXVI for 6, trifluoroacetic acid, MW, 140°C, 1 hour. 3b) Precursor PXVII for 7, trifluoroacetic acid, MW, 140°C, 1 hour.
Table 1 Histamine H₃ receptor binding data of compounds 2–7

| Compound | Structure | hH₃R affinity* $K_i$ (nM) | hH₃R hill slopes $n_H \pm$ SEM | hH₄R affinity* $K_i$ (nM) | hH₄R hill slopes $n_H \pm$ SEM | Ratio hH₄R/hH₃R |
|----------|-----------|--------------------------|-------------------------------|--------------------------|-------------------------------|------------------|
| 1        | ![Structure 1](image1.png) | 1,663±57                 | 0.672±0.001                   | 118,295±3,6905           | 0.762±0.594                   | 70               |
| 2        | ![Structure 2](image2.png) | 580±38                   | 0.796±0.104                   | 49,240±1,8424            | 0.762±0.139                   | 85               |
| 3        | ![Structure 3](image3.png) | 11.8±1.64                | 0.614±0.032                   | 14,576±6,174             | −1.026±0.03                   | 1,200            |
| 4        | ![Structure 4](image4.png) | 49.1±2.70                | 0.519±0.023                   | 4,480±79                | −1.157±0.14                   | 90               |
| 5        | ![Structure 5](image5.png) | 4.49±1.25               | 0.561±0.044                   | 29,493±2,365             | 0.423±0.019                   | 6,500            |
| 6        | ![Structure 6](image6.png) | 383±8.32                | 0.740±0.061                   | 13,245±2,795             | 1.400±0.133                   | 35               |
| 7        | ![Structure 7](image7.png) | 649±30                  | 0.803±0.047                   | 9,695±506               | 0.918±0.479                   | 15               |

Notes: [³H]N²-Methylhistamine binding assay performed with cell membrane preparation of HEK-hH₃ cells stably expressing human H₃ receptor, mean value ± SEM, $n=3$; measurement as previously described.¹⁸,¹⁹,²⁵ [³H]histamine binding assay performed with cell membrane preparation of Sf9 cells transiently expressing hH₄R and co-expressed with Gαᵢ₂ and β¹γ₂ subunits, mean value ± SEM, $n=2$; measurement as previously described.¹⁸,¹⁹,²⁵

Abbreviation: SEM, standard error of mean.

increased the affinity of the resulting compound 5 for hH₃Rs ($K_i$[hH₃R]=4.49 nM) and clearly decreased the affinity for hH₄Rs ($K_i$[hH₄R]=29,493 nM). The in-vitro observations for compound 5 at both receptor subtypes showed that proper structural derivatizations of an H₄R pharmacophore, namely 2-aminopyrimidine, can result in the design and synthesis of a highly potent and selective hH₄R antagonist with an hH₃R/hH₄R ratio of 6,500 towards hH₃Rs. However, further structural modifications of compound 5, such as replacement of basic piperidine at 2-position or N-benzylamine at 4-position of the structure, resulted in compounds 6 ($K_i$[hH₃R]=383 nM, $K_i$[hH₄R]=13,245) and 7 ($K_i$[hH₃R]=649 nM, $K_i$[hH₄R]=9,695) with a significant drop in antagonistic binding affinity towards both receptor subtypes.

Conversely, concepts of lipophilicity-related metric analyses predictively quantifying target-oriented drug-likeness have for many years been established as useful tools in the lead optimization process. Moreover, it has been shown that the lipophilicity of a respective compound is an important drug-like property, the control of which is vital for eventual success in drug development. This is not surprising since the role of LogP in altering drug potency, pharmacokinetics, and toxicity has been well established.²⁹ However, high lipophilicity is usually linked with increased likelihood of binding to multiple targets and resultant pharmacologically based toxicology.
as well as poor solubility and metabolic clearance. It has also been suggested that compounds with a clogP < 5 have a more favorable drug-likeness profile. Among the current series of compounds, clogP values were found to be < 5, suggesting their applicability for oral administration. To further assist the drug discovery/development process and to guide the optimization from a lead compound to a successful drug candidate, rules for predicting drug-like physiochemical properties have been introduced. Lipinski’s ‘Rule of five’ and other physicochemical parameters have been shown to be useful tools in selecting oral drug candidates.

Generally, values of LE and LELP being > 0.3 and < 7.5, respectively, have been validated as favorable candidates with drug-likeness. Interestingly, compounds 3, 4, and 5, the most potent hH3R antagonists among the current series, showed LE values > 0.3 for hH3R whereas a significant decrease in LE values of < 0.3 for hH4R was observed. These results further validate the drug-likeness of compounds 3, 4, and 5. Conversely, the correlation of LELP values (hH3R) for compounds 3, 4, and 5 of 9.26, 9.60, and 12.79, respectively, and 9.15, 13.11, and 23.57 for hH4R were not observed since both values were > 7.5. However, it has been proposed that a clear understanding of probabilities in drug discovery is sometimes impossible because of a significant number of known and unknown variables. As a result, metric analyses have limitations except for some

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Table 2 Molecular weight, clogP, LE, LELP, LipE, and TPSA values of compounds 2–7

| No | Structure | pKᵢ ᵃ-hH₃R | pKᵢ ᵄ-hH₄R | MWt (Å²) | cLogP | LE (Å²) ᵃ-hH₃R | LELP (Å²) ᵃ-hH₃R | LipE (Å²) ᵃ-hH₃R | TPSA (Å²) | LE (Å²) ᵄ-hH₄R | LELP (Å²) ᵄ-hH₄R | LipE (Å²) ᵄ-hH₄R |
|----|-----------|-------------|-------------|---------|-------|----------------|----------------|----------------|---------|----------------|----------------|----------------|
| 1  | ![Image](image1.png) | 5.78 | 3.93 | 348.5 | 0.44 | 76.79 | 0.32 | 1.39 | 5.34 | 0.22 | 2.04 | 3.49 |
| 2  | ![Image](image2.png) | 6.24 | 4.31 | 699.3 | 1.45 | 73.55 | 0.36 | 4.06 | 4.79 | 0.25 | 5.88 | 2.86 |
| 3  | ![Image](image3.png) | 7.93 | 4.84 | 425.3 | 3.25 | 82.78 | 0.35 | 9.26 | 4.68 | 0.21 | 15.17 | 1.59 |
| 4  | ![Image](image4.png) | 7.31 | 5.35 | 439.3 | 3.01 | 82.78 | 0.31 | 9.60 | 4.30 | 0.23 | 13.11 | 2.34 |
| 5  | ![Image](image5.png) | 8.35 | 4.53 | 533.3 | 4.73 | 62.31 | 0.37 | 12.79 | 3.62 | 0.20 | 23.57 | −0.20 |
| 6  | ![Image](image6.png) | 6.42 | 4.88 | 780.8 | 3.73 | 65.55 | 0.28 | 13.55 | 2.69 | 0.21 | 17.82 | 1.15 |
| 7  | ![Image](image7.png) | 6.19 | 5.01 | 425.6 | 3.49 | 60.00 | 0.27 | 9.09 | 3.70 | 0.22 | 11.21 | 2.52 |

Notes: [¹³]N=ethylhistamine binding assay performed with cell membrane preparation of HEK-293 cells stably expressing hH₃R, n=3; [³]H]histamine binding assay performed with cell membrane preparation of SF9 cells transiently expressing hH₄R and co-expressed with Gαi2 and β1γ2 subunits, n=2; calculated using MarvinSketch 6.0.5 and ChemAxon (demo); calculated as previously described.26–39
metrics that can be useful such as LipE and enthalpy in drug provability.\textsuperscript{38,39} It is therefore important to know when metric-based information is useful, useless, or possibly misleading so that it can be appropriately prioritized.\textsuperscript{26} Interestingly, LipE values, which are considered to be independent of molecular size of the respective compound, were found to be <5 for the three most promising compounds based on both histamine receptor subtypes. LipE results further corroborated the drug-likeness of compounds 3, 4, and 5. Moreover, it has been previously suggested that compounds with TPSA values >60 Å\textsuperscript{2} are generally regarded as poor membrane-permeable substances with predictively reduced CNS bioavailability.\textsuperscript{36,37} Among the synthesized series, the observed TPSA values for compounds 2–7 were in the range of 60 Å\textsuperscript{2}–82 Å\textsuperscript{2}, necessitating further structural optimization of the current class with the objective of generating newer derivatives with enhanced predictive physicochemical parameters, especially for TPSA. In this regard however, the results for compounds 2–7 showed that an increase in size and surface improved affinity and selectivity for hH\textsubscript{R} was not successful concerning drug-likeness.

Conclusion
With respect to the development of potential histamine hH\textsubscript{R} ligands, results observed for compounds 2–7 indicated that slight structural changes evoked extensive differences in affinities as well as histamine receptor subtype preferences. The incorporation of a piperidin-1-ylpropoxyphenyl, an established hH\textsubscript{R} pharmacophore, in the hH\textsubscript{R}-affine structural scaffold 2-aminopyrimidine increased hH\textsubscript{R} affinity and subtype selectivity while hH\textsubscript{4}R affinity was decreased. Assuming reference compound 1 as starting lead, some spacer variations as well as an exchange of the lipophilic 4-methylpiperazine residue by N-benzylamine and a combination of both were performed. Compounds 3, 4, and 5 showed reduced affinities at hH\textsubscript{R} with concentration ranges significantly lower than that of reference compound 1. These results indicate that incorporation of an H\textsubscript{R} pharmacophore can provide greater affinity and selectivity for hH\textsubscript{R} over hH\textsubscript{R}. For example, the derivative with a 3-piperidinopropyloxy-substituted amine group at 2-position (compound 5) was 6,500× more selective for hH\textsubscript{R} than hH\textsubscript{R}, with further corroboration of these results provided by physicochemical and drug-likeness parameters.

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Disclosure
The authors declare no conflicts of interest in this work.

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