SilE is an intrinsically disordered periplasmic “molecular sponge” involved in bacterial silver resistance

Introduction

Silver is a soft, shiny, lustrous and precious metal (Lansdown, 2010) with high value as a human commodity in jewellery and as an investment, and with wide applications in the electronics industry; approximately 24,000 tons of silver was mined and produced in 2012 (Mijndendonckx et al., 2013). Silver has also been highly valued for its broad-spectrum antimicrobial properties and has been one of the most important antimicrobial agents prior to the discovery and introduction of antibiotics. The rapid emergence of antibiotic resistance among many bacteria has rejuvenated the interest in silver as a viable alternative antimicrobial agent (Holt and Bard, 2005; Atiyeh et al., 2007; Mijndendonckx et al., 2013).

The widespread use of silver in medical and non-medical settings has resulted in the emergence of silver resistant bacteria. Initial reports of silver resistance date back to 1966 (Gupta et al., 1999; Mallard et al., 2012) and the first silver-resistant plasmid pMG101, a large 180 kb plasmid assigned to the IncHI incompatibility group and reportedly carrying the sil operon (conferring silver resistance), was isolated from Salmonella enterica serovar Typhimurium following the death from septicaemia of several patients treated with silver nitrate, leading to the closure of the burn ward of the Massachusetts General Hospital (McHugh et al., 1975). The pMG101 sil resistance allowed growth of an Escherichia coli K-12 (E. coli) strain carrying pMG101, in standard Luria-Bertani (LB) broth containing 600 μM of Ag+, a concentration over six times of that known to be tolerated by E. coli strains K-12 strains lacking the plasmid (Gupta et al., 1999).

However, the sil operon comprises nine genes sil-PGABFCRSE (Silver, 2003; Mijndendonckx et al., 2013; Hobman and Crossman, 2015; Randall et al., 2015), organised into three transcriptional units, silCFBAGP, silRS and silE, each controlled by a different promoter (Silver, 2003). The corresponding proteins have been assigned putative roles based upon homology modelling compared with other known heavy metal resistant determinants of the Pco or Cus systems (Hobman and Crossman, 2015 and Fig. 1). silE is under the control of its own promoter and its expression is strongly induced by Ag+...
Fig. 1. Silver resistance operon and functions of its corresponding proteins.

Silver resistance proteins and their suggested active roles, deduced from homology, thought to partake in bacterial silver resistance. SilE—periplasmic metal-binding protein, SilR and SilS—responder and membrane sensor performing two-component transcription regulation, SilC—outer membrane protein, SilB—membrane fusion protein, SilA—chemi-osmotic antiporter, SilP—P-type cation ATPase and SilG (protein not depicted) and SilF—metal-binding protein, SilA—chemi-osmotic antiporter, SilP—P-type cation ATPase and SilG (protein not depicted) and SilF—metal-binding protein, SilA—chemi-osmotic antiporter, SilP—P-type cation ATPase and SilG (protein not depicted) and SilF—metal-binding protein, SilA—chemi-osmotic antiporter, SilP—P-type cation ATPase and SilG (protein not depicted) and SilF—metal-binding protein. Dashed arrows highlight the hypothetical role of SilE. The bottom line shows the mRNAs, indicating the genes and open reading frames (including number of amino acids) with the orientation of their transcription (Silver, 2003).

In this paper, we provide experimental evidence that apo-SilE is an intrinsically disordered protein (IDP) that folds to a highly α-helical holoSile structure upon binding to Ag⁺ ions. SilE can bind up to eight Ag⁺ ions or fewer of the harder divalent metal ions Cu²⁺ (up to six), Zn²⁺ (up to five) and Ni²⁺ (up to two), indicating a higher capacity for complexing Ag⁺ compared to other metals. We show that metal-induced folding leads to a higher helical content with Ag⁺ followed by Cu²⁺, Zn²⁺ and Ni²⁺, consistent with its higher selectivity for Ag⁺, and confirm from mutagenesis studies that conserved histidine and methionine residues within specific sequence motifs are involved in Ag⁺ binding. We propose an Ag⁺-mediated nucleation folding mechanism for SilE and suggest that SilE acts as a “molecular sponge” and as a first line of defence against relatively low levels of Ag⁺ ions that enter the periplasm. Potential consequences of Ag⁺-mediated SilE folding relative to the combined bacterial silver resistance mechanism are also discussed.

Results

Significant primary sequence and structural features of SilE

Sequence alignment of native SilE and its Cu-binding homologue PcoE was carried out using a general multi-purpose primary sequence alignment program for proteins Omega (McWilliam et al., 2013), and amino acid sequences were coloured according to the “Percentage Identity” between the two proteins in Jalview (Waterhouse et al., 2009), (Fig. 2A). The calculated sequence identity between the two homologue proteins was 48% (Zimmerman et al., 2012). SilE and PcoE are rich in potential metal ligand-binding histidines (ten each) and methionines (ten and fifteen, respectively). The primary sequence identity shows that the position of the ten histidine residues of the two proteins is completely conserved (Fig. 2A), suggesting a key role in metal binding.
SilE has been predicted to be an IDP. The SilE sequence was analysed through DisEMBL’s and the output indicated 56% loop/coil (Supporting Information Fig. S1), suggesting that SilE has a high proportion of intrinsic disorder. In parallel, screening for possible regions of secondary structure (α-helical, β-strand or random coil) within the SilE (using Jpred, Cole et al., 2008) predicted six α-helices across 54% of the protein sequence (Fig. 2B). Interestingly, the secondary structure predictions through Jpred are consistent with the far-UV CD data obtained for the holo-form of SilE (Supporting Information Table S1). Furthermore, two characteristic motifs, MxxHxxxxxHxxMxx (motif 1) and HxxMxxxHxxMxx (motif 2), each repeated twice within the sequence have conserved histidines and methionines that constitute potential metal-binding motifs (Fig. 2C).

Structural analysis of SilE with and without Ag⁺ by CD and NMR

The secondary structure of SilE was determined by far-UV circular dichroism (CD) spectroscopy (Fig. 3A). With strong negative signals around 200 nm, the spectrum obtained for apo-SilE in 10 mM HEPES, 20 mM NaF, pH 7.5, is typical of an unstructured, random coil polypeptide. Slight negative shoulders on the CD spectrum, at 207 and 221 nm, are consistent with a minor fraction (<20%) of α-helical secondary structure for apo-SilE (Sreerama et al., 1999; Greenfield, 2007; Dodero et al., 2011). However, these bands become considerably more prominent (~54%) when bound to Ag⁺ in the holo-SilE, with a strong negative band at 207 nm, a weaker negative ellipticity at 221 nm and a strong positive band at 190 nm (Fig. 3A), consistent with the stabilization of helical structure.

NMR spectroscopy was used to examine the tertiary structure with and without Ag⁺ (in 10 mM HEPES, 20 mM NaF, pH 7.5). The two-dimensional 1H-15N HSQC of apo-SilE (Fig. 3B) exhibited poorly dispersed peaks, characteristic of an unfolded and unstructured protein (Dyson et al., 2005), and provided further experimental evidence for the largely disordered and flexible nature of SilE under native conditions. Likewise, one-dimensional proton NMR experiments (in 10 mM sodium phosphate buffer) at several pH intervals between pH 9 and pH 5, showed no change in the local environment of apo-SilE (data not shown). However, the dispersion within the NMR backbone amide 1H chemical shifts in the 1H-15N HSQC spectrum of holo-SilE (Fig. 3B) increased substantially in the presence of bound Ag⁺ showing a clear signature for the induction of hydrogen bonded secondary structure.
We investigated the thermal stability of the Ag-bound SilE complex using CD by generating a melting curve and monitoring the change in ellipticity at 207 nm. (Fig. 3C). The unfolding showed a sigmoidal transition with a mid-point of 42°C. Moreover, the relatively sharp transition from the folded to unfolded form, (Honig et al., 2003), as the temperature increased suggested that only two conformational states are significantly populated.

It appears that SilE folding upon binding to Ag occurs cooperatively, as indicated by the sigmoidal thermal stability curve deduced via far-UV CD (Fig. 3C). The monophasic, cooperative unfolding of the protein confirmed that with Ag present, the protein exists as a compact well-folded, stable structure up to a temperature of 42°C. Moreover, the relatively sharp transition from the folded to unfolded form, characteristic of two-state proteins (Horng et al., 2003), as the temperature increased suggested that only these two conformational states were present to any significant extent.

SilE binds up to eight silver ions

Interactions of certain metal ions with SilE have been reported using nano-ESI-MS (nano-Electrospray Ionization-Mass Spectrometry) under non-denaturing conditions in volatile 25 mM ammonium acetate buffer at pH 7.0. Solutions of 2 mM of each metal ion: Ag⁺, Cu²⁺, Zn²⁺ and Ni²⁺ were added to 25 µM apo-SilE (molar ratio of 80:1). SilE-metal complexes were observed, in each case with the relative bound proportions dependent on the protein’s affinity and/or stoichiometry for the metal ion (Fig. 4A). SilE shows a distribution of Ag-bound complexes with species containing 5, 6 and 7 bound Ag⁺ ions particularly abundant, with a maximum number of 8 detected. Binding of other harder divalent metal ions Cu²⁺ and Zn²⁺ was also evident, but with different binding stoichiometries (SilE:Cu²⁺ of 1:6; SilE:Zn²⁺ of 1:5, and a lower stoichiometry with SilE:Ni²⁺ of 1:2).

Complementary titrations using far-UV CD (Fig. 4C) show no further change in secondary structure content after the addition of 6 molar equivalents of Ag⁺. Hence six Ag⁺ appears to be the optimum for full folding of the protein, however, the MS data suggest a further two Ag⁺ ions are capable of being bound by the folded protein. The Ag⁺ CD titration shows a very clear isodichroic point around 206 nm, which is consistent with predominantly two species in solution, namely the apo-SilE and a predominant single ‘fully loaded’ Ag⁺-bound form, rather than a heterogeneous mixture of different species with different binding stoichiometries. Furthermore, plots of the CD signals at 190 and 221 nm as a function of the Ag⁺/SilE molar ratio show some evidence for sigmoidal curves consistent with co-operative uptake of Ag⁺ by SilE (Fig. 4D).
Histidine and methionine residues are involved in Ag⁺-binding

In order to investigate the roles of conserved histidine and methionine residues in Ag⁺-binding and protein folding, we mutated nine histidines to alanines (H38A, H62A, H69A, H80A, H87A, H111A, H118A, H129A and H136A) and four methionines to leucines (M72L, M83L, M90L and M108L). The secondary structure contents of all the mutant SilE proteins were measured in the presence of six Ag⁺ molar equivalents (Fig. 5A–C). H38A, H62A, H69A, H118A, H129A and H136A gave only a slight decrease in α-helical structure (Fig. 5A and Supporting Information Table S1). In contrast, H80A, H87A and H111A exhibited large reductions in α-helical content in comparison to wild-type holo-SilE (Fig. 5A and Supporting Information Table S1), implying that these residues are important for Ag⁺-induced folding. The largest change in the CD spectra was seen with H111A which exhibits only around 60% of the wild type holo SilE secondary structure content (Supporting Information Table S1).

Based on this data, we then made three double mutant proteins, H80A/H87A, H80A/H111A and H87A/
H111A. Far-UV CD spectra of these proteins (Fig. 5B) showed a decrease in secondary structure content greater than that found with the single histidine mutant proteins with \(\alpha\)-helical content declining to half that of the wild type. Similar measurements for the methionine mutants found that the M108L mutant had a comparable decrease in secondary structure content to the H111A protein (Fig. 5C). The other three methionine mutants (M90L, M83L and M72L) showed effects similar to those observed for the histidine mutants (compare Fig. 5A and C).

We then measured the number of Ag\(^{+}\) bound to each of the SilE proteins using inductively coupled plasma mass spectrometry (ICP-MS). The single histidine mutants H62A, H69A, H80A, H87A, H111A, H118A and H129A all bound on average one Ag\(^{+}\) less than the native holo-SilE protein (Fig. 5D). The other three methionine mutants (M90L, M83L and M72L) showed effects similar to those observed for the histidine mutants (compare Fig. 5A and C).

Two 'core-motifs' central to the primary sequence of SilE likely form the nucleation site for Ag\(^{+}\)-induced folding

From our studies of mutant SilE proteins it appears that the H80A, H87A, H111A and M108L exhibited the largest folding defects upon Ag\(^{+}\) binding. These residues are located centrally within the sequence in an apparent core comprising a motif 1 (residues A77-M91) and a motif 2 (residues E110-F120) (Fig. 2B). Given that the secondary structure content has a high sensitivity to their mutation, we proposed that they provide initial nucleation sites for Ag\(^{+}\)-induced folding. To test this hypothesis we engineered two truncated SilE polypeptides, one SilE\(_{46-128}\) that preserves the core region (P46 to P128) but lacks the peripheral sequences at the N- and C-terminal regions, and a second SilE\(_{21-98}\) that lacks the C-terminal region and motif 2 from the putative nucleation core (Fig. 6A). We then studied Ag\(^{+}\)-mediated folding of these polypeptides using far-UV CD (Fig. 6B). The SilE\(_{46-128}\) polypeptide exhibited Ag\(^{+}\)-mediated folding similar to the wt SilE whereas the SilE\(_{21-98}\) polypeptide did not exhibit any signs of folding upon Ag\(^{+}\) binding. These data are consistent with a model where a nucleation core is formed by central core motifs 1 and 2 which then facilitates further folding as more Ag\(^{+}\) are bound to the rest of the polypeptide.
metals Cu\(^{2+}\) and Ag\(^{+}\) (Kosol et al., 2013; Sibile and Bernado, 2012; Jensen et al., 2013). Previous NMR studies of the SilE homologue protein PcoE have shown that it folds to a predominantly $\alpha$-helical structure upon binding ligand-metals Cu\(^{2+}\) and Ag\(^{+}\) (Zimmerman et al., 2012). Likewise, SilE is thought to be a putative IDP that binds to Ag\(^{+}\) but no experimental evidence was previously available to verify its structure and metal binding properties. The data presented in this paper show that that apo-SilEs is an IDP lacking significant secondary structure over a range of pH conditions (Fig. 3). However, we were then able to demonstrate strong coupling between folding and metal-uptake by showing that SilE binds Ag\(^{+}\) (Figs. 3 and 4) and folds into a moderately stable structure of high $\alpha$-helical content. Despite the involvement of multiple histidine residues in Ag\(^{+}\) binding, no pH induced folding of SilE was observed, verifying that SilE folding is specifically mediated by Ag\(^{+}\) binding.

Each SilE molecule can bind up to a maximum of eight monovalent soft metal ions of Ag\(^{+}\) in solution (Fig. 4A) but is fully folded after binding six Ag\(^{+}\): the last two Ag\(^{+}\) ions must therefore bind only to the fully folded protein. Cu\(^{2+}\), Zn\(^{2+}\) and Ni\(^{2+}\) ions, exhibit lower stoichio-

Discussion

Many proteins are intrinsically unstructured under physiological conditions, yet can fold when required to perform their biological functions (Dyson et al., 2005; Radivojac et al., 2007; Sibile and Bernado, 2012; Jensen et al., 2013; Kosol et al., 2013). Previous NMR studies of the SilE homologue protein PcoE have shown that it folds to a predominantly $\alpha$-helical structure upon binding ligand-metals Cu\(^{2+}\) and Ag\(^{+}\) (Zimmerman et al., 2012). Likewise, SilE is thought to be a putative IDP that binds to Ag\(^{+}\) but no experimental evidence was previously available to verify its structure and metal binding properties. The data presented in this paper show that that apo-SilEs is an IDP lacking significant secondary structure over a range of pH conditions (Fig. 3). However, we were then able to demonstrate strong coupling between folding and metal-uptake by showing that SilE binds Ag\(^{+}\) (Figs. 3 and 4) and folds into a moderately stable structure of high $\alpha$-helical content. Despite the involvement of multiple histidine residues in Ag\(^{+}\) binding, no pH induced folding of SilE was observed, verifying that SilE folding is specifically mediated by Ag\(^{+}\) binding.

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This use of folding and binding to enable uptake of a specific ion gives SilE the functional role of a ‘molecular sponge’ in the mechanism of E. coli silver resistance. As such its relationship to the other components in the silver resistance mechanism must be one where the unfolded protein has a high affinity for silver, but a low affinity for other cellular components. In contrast the folded protein will have a high affinity for other cellular components and as such allow SilE to off-load its Ag\(^{+}\) to other components in the resistance mechanism. Accordingly, SilE may act either as a metal-ion chaperone to the RND effluxer SilCBA or, as many IDPs do (Kosol et al., 2013) binds to or signals to the histidine kinase sensor SilS and thereby activates the remainder of the silver resistance machinery. Therefore, the the notion that SilE binds Ag\(^{+}\) and then initiates the entire mechanism of silver resistance (Sil proteins P, G, A, B, F, C, R and S, as well as positively autoregulating its own expression), by signalling or direct binding to SilS, in bacteria is a highly plausible hypothesis.

Primary sequence alignment between SilE and its homologue PcoE revealed conserved histidine and methionine residues (Fig. 2) (Mirolo et al., 2012) which in $\alpha$-helical structures are known to be a common feature of many metal-binding proteins (Todd et al., 1991; Tanaka et al., 2004). We have further confirmed this to be the case in SilE by site-specific mutagenesis. Importantly, our mutagenesis studies showed that some residues—H80, H87, H111 and M108—play a much more significant role than others in the correct Ag\(^{+}\)-mediated folding of SilE (Fig. 5). Secondary structure predictions were consistent with our experimental data, suggesting the presence of six $\alpha$-helices, however, significant helical structure was only realised in the Ag-bound state (Fig. 2). In addition, two types of Ag\(^{+}\)-binding motifs, motif 1 MxxHxxxxxxHxxMxx and motif 2 HxxMxxxHxxMxx, where identified from the sequence in which $i$ to $i+3$ or $i+4$ spacing between residues ensures that they appear on the same face of a folded helix. This suggests that each single helical motif in itself is unable to bind Ag\(^{+}\), but several of these together are able to (co-operatively) co-ordinate multiple Ag\(^{+}\) ions leading to a compact protein fold. Two of these motifs are located at the centre of the primary SilE sequence between residues S76-G124 (Fig. 2) and we define them as the “core-motifs” (Fig. 6). Both H80 and H87 are located in the core-motif 2 whereas M108 and H111 are located in the core-motif 1. We suggest the possibility of an Ag\(^{+}\)-mediated folding mechanism whereby initial Ag\(^{+}\) binding to these “core-motifs” induces a
nucleation site around which the rest of the SilE structure assembles facilitating further Ag$^+$ binding (Fig. 6). This is consistent with the sigmoidal plots of CD signals at 190 and 221 nm as a function of the Ag$^+$/SilE molar ratio (Fig. 4D) which are consistent co-operative Ag$^+$ binding and a folding nucleus rather than a simple sequential Ag$^+$ binding and folding model. We tested this model by engineering truncated SilE polypeptides. The SilE$^{16-127}$ polypeptide contained both “core motifs” and exhibited Ag$^+$-mediated folding in a similar manner to the native SilE whereas the SilE$^{21-97}$ polypeptide lacking the core-motif 1 exhibited no detectable Ag$^+$-mediated folding: a result consistent with our nucleation model. Our double histidine mutations H80A/H87, H87A/H111A and H87A/H111A revealed loss of two Ag$^+$ ions and it is therefore likely that these pairs of histidines coordinate different Ag$^+$ ions.

Complexes of Ag$^+$ with histidine and imidazole can form with Ag$^+$$\cdot$$L$ and Ag$^+$$\cdot$$L_2$ stoichiometries (where L is the ligand; histidine or imidazole) with the latter being more stable, as indicated by higher stability constants and enthalpies of formation (Czoik et al., 2003). Methionine residues are reported to coordinate Ag$^+$ ions (Du et al., 2007). The human copper transporter 1 (hCtr1) protein in turn co-ordinates Cu$^+$ ions by metal binding processes or using a Vivaspin device. The metal ion salts used are as follows: Ag$^+$ from silver nitrate (AgNO$_3$), Cu$^{2+}$ from cupric sulfate (CuSO$_4$), (Zn$^{2+}$) from zinc chloride (ZnCl$_2$) and Ni$^{2+}$ from nickel chloride (NiCl$_2$).

Construction of expression plasmid

The DNA sequence of the gene encoding SilE, minus the 20-amino acid peptide leader sequence, was amplified by PCR from the E. coli plasmid pMG101 (Gupta et al., 1999). The two primer sequences are: for the forward primer 5'-ACTGAAACCGTGAATATCCATG-3' and for the reverse primer 5'-GCCTGACGTGACATGCG-3'. To facilitate DNA cloning and protein expression, an Opti3Clinfwd site was incorporated in the forward primer and in the reverse primer, an Infusion 3' site including a stop codon were integrated. The PCR product was then built into the expression vector pOPINF (OPPF), comprising the coding sequence of a His-tag, by In-Fusion Reactions (Bird, 2011). The expression construct was validated to contain the correct gene sequence insert by PCR screening, using the same cloning primers.

PCR-based site-directed mutagenesis

The SilE-pOPINF construct served as a template for PCR-based site-directed mutagenesis. Histidine to alanine (single and double) mutations as well as the methionine to leucine mutations were generated in 5' and 3' DNA fragments using the following primer combinations, where mutant introducing nucleotides are shown in bold, lowercase. To facilitate DNA cloning and protein expression, an Opti3Clinfwd site was included. The PCR product was then built into the expression vector pOPINF (OPPF), comprising the coding sequence of a His-tag, by In-Fusion Reactions (Bird, 2011). The expression construct was validated to contain the correct gene sequence insert by PCR screening, using the same cloning primers.

Experimental procedures

Materials

Chemicals and reagents (analytical grade) were obtained from Sigma Chemical, unless otherwise stated. All buffer exchanges were completed either by centrifugal ultrafiltration through a high flux polyethersulphone membrane with molecular weight cut-off at 5,000 Da on a Vivasin 20 or 2 devices or, by three cycles of four-eight hours of dialysis against the new buffer at 4°C, using dialysis membranes with molecular weight cut-off at 3,500 Da (Spectrum Laboratories Inc.). The sources of the metal ions were always soluble and excess ions were removed via extensive dialysis or using a Vivaspin device. The metal ion salts used are as follows: Ag$^+$ from silver nitrate (AgNO$_3$), Cu$^{2+}$ from cupric sulfate (CuSO$_4$), (Zn$^{2+}$) from zinc chloride (ZnCl$_2$) and Ni$^{2+}$ from nickel chloride (NiCl$_2$).

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| SilE 3' Fwd   | SilE 3' Rev   |
|--------------|--------------|
| GGCACCTGTGCCGACATGGA-3' | GCAGGATGGAATTCATACATG-5' |
| ACTGCATCTGggcAGCGAGTTGCC-3' | 5'-TATGACACACggcGAACCCGACTATCTTCATGGCC-3' |
| 5'-GCCTGACGTGACATGCG-3' | 5'-CACTGAGCAATATGGCCTGTTGggcc-3' |
| 5'-GGCAGGATGCGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GCCTGACGTGACATGCG-3' |
| 5'-GGCAGGATGCGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
| 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' | 5'-GGCTGAAACTTGggccAGAATGGAAGAGCTCC-3' |
silEH118AFwd; 5′—TGCAGTTGCcgcGAATTATGAA-
TAACG—3′
silEH118ARev; 5′—GGTTATCTAATAATCTGGcGGCAAC-
TGCA—3′
silEH129AFwd; 5′—GTCTGGAACGCcCAGCCCATG-
G—3′
silEH129ARev; 5′—CCATGGCCTGGcGCCAGA-
C—3′
silEH136AFwd; 5′—GCGCAAGCGGCCGTCGCA-
TAG—3′
silEH136ARev—5′—GCATGCAGACGgcccGCTTGCGC-
C—3′
silEM72LFwd; 5′—TCATGAAACGcCtGACCGGGTC—3′
silEM72LRv; 5′—GACCCGGTTCGcagGTTTCTAGA—3′
silEM83LFwd; 5′—GCACCAAGAactgGTGGAAGCTCA-
TAG—3′
silEM83LRv; 5′—CTGATGACTTTCCACcagTTTCTGG-
TG C—3′
silEM90LFwd; 5′—TCATCGAGGAgCtGATGGGAAGTCA-
GAC—3′
silEM90LRv; 5′—GCTGTCCCTCATAcGTCTGAT-
GA—3′
silEM108LFwd; 5′—ATTAGCCGCaAgtCATTAGACCAT-
GA—3′
silEM108LRv; 5′—CTTTATGCCTCATcagTGCCTGG-
CTT—3′
silePpuMIF; 5′—ATCCCCGGAAGTTAATCCggaccTATA-
TTA C—3′
sileHindIII R; 5′—ATCACAAACTCTGCTAAGcTattGAC-
CTGC—3′

The 5′ and 3′ SilE mutated fragments were used as mega primers in a PCR including the flanking primers to generate the cDNA containing the entire translated region of the mutated His-tagged SilE. The final cDNA constructs were cloned into the PpuMI/HindIII site, of the prokaryotic expression vector pOPINF. The introduction of the mutations as well as the absence of undesired spontaneous mutations was confirmed by sequencing.

Engineering of SilE truncated polypeptides

The SilE-pOPINF construct served as a template for PCR-based production of the truncated polypeptides SilE46-128 and SilE21-98, using the following primers –

silE46-128 F; 5′—ATCCCCCGAGGAAGTTAATCCggaccTATA-
TTATTC—3′
silE46-128 R; 5′—GCTATGAAAGCCTTCGGAATTAAG-
GGAAACG—3′
silE21-98 F; 5′—CAATCGGCGGCCCAGCCTGCGGATC-
CAGGG—3′
silE21-98 R; 5′—GCCTCAGAAGCCTTATAGGGcCAG-
CAG—3′

The final cDNA constructs were cloned into the PpuMI/HindIII site for SilE21-98 and into the Apai/HindIII site for SilE46-128, of the prokaryotic expression vector pOPINF, ensuring the presence of the N-terminal His-tag remained. Confirmation of complete cloning of truncated polypeptides as well as the absence of undesired spontaneous mutations was confirmed by sequencing.

Protein overexpression and purification

All SilE over-expression plasmids (wild-type and mutants) were maintained in E. coli DH5α cells and transformed into E. coli BL21 Star (DE3) cells (Invitrogen) for SilE wild-type and mutant over-expressions. Unlabelled samples of SilE were produced from cells grown on LB medium. 15N labelled SilE was prepared in standard minimal media supplemented with 15N-H4Cl. Each litre of medium, supplemented with 50 mg carbenicillin or 100 mg ampicillin, was inoculated with an overnight culture (5-10 mL) of the transformed E. coli cells. The cells were grown aerobically with vigorous shaking at 200 rpm, 37°C, to OD600 ~0.6-0.8 and isopropyl β-D-1-thiogalactopyranoside (IPTG) was added at a final concentration of 1mM to induce protein expression. After overnight induction at 37°C (200 rpm) the cells were harvested by centrifugation prior to lysis by sonication (Soniprep 150) at an output frequency of 23 kHz 12 cycles of 20 seconds at an amplitude of 10 microns, followed by 30 seconds of recovery were carried out in buffer containing 50 mM Tris- (tris(hydroxymethyl)aminoethane) HCl, 500 mM NaCl, pH 7.5, supplemented with 100 μg/mL lysozyme and 1 mL per 20 g cells of protease inhibitor cocktail (Sigma; for use in purification of His-tagged proteins, DMSO solution). The cell lysate was clarified by centrifugation and the supernatant contained the soluble SilE protein.

All SilE proteins were expressed with an N-terminal His-tag to allow purification by affinity chromatography on nickel-chelating Sepharose (GE Healthcare) (in column binding buffer 50 mM Tris-HCl, 500 mM NaCl, pH 7.5 with elution in 500 mM imidazole), after which they were further purified by size-exclusion chromatography (SEC) on a pre-equilibrated (20 mM Tris-HCl, 200 mM NaCl, pH 7.5) Superdex 75 column (10 mm x 300 mm) (GE Healthcare). The His-tag was then cleaved off by incubation at 4°C overnight, with 2.5 μg HRV 3C protease per 10 mL of protein sample and all forms of SilE finally were further purified by a second round of affinity chromatography on nickel-chelating Sepharose (GE Healthcare), whereby the protein was eluted in 50 mM Tris-HCl, 500 mM NaCl, pH 7.5 with a final concentration imidazole (with the histidine mutants and truncated polypeptides requiring a lower concentration imidazole), pH 7.5. The purity and identity of the SilE proteins was confirmed by; SDS-PAGE and nano-ESI MS, which yielded a molecular mass of 13,268 (+/-1.6) Da for wild-type (Suppl. Fig. S2), corresponding to the values calculated from the sequences of SilE. All the purified SilE proteins contained no detectable metals.

Protein Concentrations

SilE lacks light absorbing tryptophan tyrosine residues and its concentration was estimated via two alternative methods. Firstly, acquiring the Brix coefficient using an Atago DD-7 Digital Differential Refractometer allowed us to calculate the protein concentration using the following formula as explained elsewhere (Theisen et al., 2000):

\[
\text{protein concentration} = \frac{(dnde sucrose/dnde protein sample) \times 7.8883}{x Brix coefficient} = (0.15/0.18) \times 7.8883 \times \text{Brix coefficient}
\]

Secondly, by measuring the absorbance of the peptide bond at the ultraviolet wavelength of 205 nm on a Thermo
Circular dichroism spectroscopy

Far-UV CD was used to determine the secondary structure of apo- and holo-SiIE in solution. CD experiments were conducted at CD1 beam line at the ASTRID2 storage ring facility at Aarhus University, Aarhus, Denmark (Hertela and Hoffman, 2011). Data was acquired at 25°C from 50 μg apo- and metal ligand ion (Ag⁺, Cu²⁺, Zn²⁺, Ni²⁺) bound-SiIE protein samples at 57 or 75 μM in a quartz suprasil cylindrical cell (Hellma type 121.000) in a 10 mM HEPES (4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid), 20 mM NaF buffer at pH 7.5, with either 2 mM or titrated quantities (1-16 equiv. Ag⁺) of metal ions added to the holo samples. Spectra were recorded from 170 to 280 nm, with the protein sample being in a 0.05 cm path length cell; scan speed of 20 nm min⁻¹ and a response time of 1 s, with each spectrum representing an average of three accumulations, with an average of 15 scans per point. A scan of buffer alone was subtracted from the protein curve. Data were converted to molar CD per residue and spectra analysis was carried out by comparing the profile of the obtained curve to those illustrated and quantified in literature (Sreerama et al., 1999; Greenfield et al., 2007; Dodero et al., 2011). Secondary structure percentages were calculated using the DichroWeb (Lobley et al., 2002) interfaces analysis programme CONTINLL, which implements the locally linearised algorithm in selecting protein sets from the reference database (Provencher and Glockner, 1981; Van Stokkum et al., 1990; Seerama and Woody, 2000).

The stability of apo and holo-SiIE (2 mM Ag⁺) to temperature denaturation was tested and determined by following changes in the CD spectra. The changes in the intensity of the maximal negative signal at 200 nm for apo and positive signal at 190 nm for holo (2 mM Ag⁺) were recorded as a function of increasing temperature from 18 to 80°C. The temperature was gradually increased at increments of 1°C per minute and protein samples were allowed to equilibrate at each temperature, prior to recordings at intervals of 5°C. In each case, spectra were acquired from 100 μl protein samples at 57 μM. The CD data was converted to a percentage change of the maximum CD (mdeg).

Nano-electrospray ionisation mass spectrometry

Experiments were carried out and spectra were recorded on a SYNAPHT High Definition Mass Spectrometry (HDMS) (Waters) a hybrid quadrupole ion mobility time-of-flight MS instrument, with travelling-wave ion mobility (TWIM) capability, equipped with the standard z-spray source. The instrument conditions were optimized to provide the highest relative signals for apo- and 2 mM metal ligand ion (Ag⁺, Cu²⁺, Zn²⁺, Ni²⁺) bound-SiIE complexes at a protein concentration of 25 μM, sprayed from 25 mM ammonium acetate (C₂H₃NO₂) pH 7.0. The nano-ESI capillary voltage was 1.5 kV; cone voltage, 20 V; extraction voltage, 5 V; transfer voltage, 5 V. Other settings were as follows: trap and transfer collision voltage, 6 and 5 V, respectively; backing pressure, 1.6–1.8 mbar; trap pressure, 2.1 × 10⁻² mbar; TOF region pressure, 1.5 × 10⁻⁶ mbar. Instrument control as well as data processing was carried out with the Waters MassLynx 4.1 data system. All spectra were acquired in ion positive mode and the TOF analyser operated on V-mode. Minimum smoothing and background subtraction was applied to the obtained spectra prior to analysis.

ICP-MS

Ag⁺ elemental analysis of 50 nM protein diluted, with and without Ag⁺ in 5 mL solutions of 10 mM HEPES, 20 mM NaF, pH 7.5, in 1% HNO₃, was undertaken by ICP-MS (Thermo-Fisher Scientific iCAP-Q; Thermo Fisher Scientific, Bremen, Germany). The instrument was run employing three operational modes, including (i) a collision-cell (Q cell) using He with kinetic energy discrimination (He-cell) to remove polyatomic interferences, (ii) standard mode (STD) in which the collision cell is evacuated and (iii) hydrogen mode (H₂-cell) in which H₂ gas is used as the cell gas. Samples were introduced from an autosampler (Cetac ASX-520) incorporating an ASXpress™ rapid uptake module through a PEEK nebulizer (Burgher Mira Mist). An internal standard Rh (10 μg L⁻¹) in 2% trace analysis grade (Fisher Scientific, UK) HNO₃ was introduced to the sample stream on a separate line via the ASXpress unit. External multi-element calibration standards (Claritas-PPT grade CLMS-2 from SPEX Certiprep Inc., Metuchen, NJ) included Ag, Al, As, Ba, Be, Cd, Ca, Co, Cr, Cs, Cu, Fe, K, Li, Mg, Mn, Mo, Na, Ni, P, Pb, Rb, S, Se, Sr, Ti, U, V and Zn, in the range 0–100 μg L⁻¹ (0, 20, 40, 100 μg L⁻¹). A bespoke external multi-element calibration solution (PlasmaCAL, SCP Science, France) was used to create Ca, Mg, Na and K standards in the range 0–30 mg L⁻¹. Phosphorus, boron and sulphur calibration utilized in-house standard solutions (KH₂PO₄, K₂SO₄ and H₃BO₃). In-sample switching was used to measure B and P in STD mode, Se in H₂-cell mode and all other elements in He-cell mode. Peak dwell times were 10 ms for the element with 150 scans per sample. Sample processing was undertaken using Qtegra™ software (Thermo-Fisher Scientific) utilizing external cross-calibration between pulse-counting and analogue detector modes when required with data being acquired in μg L⁻¹. Protein concentrations were measured before and after the experiment. All glassware and plasticware used for these experiments were washed with 10% nitric acid to remove contaminating metal.

Nuclear magnetic resonance spectroscopy

One-dimensional ¹H nuclear magnetic resonance (NMR) spectra were taken at 25°C on an 800 MHz Bruker Avance NMR machine, using 500 μM protein in 10 mM sodium phosphate (Na₃PO₄/NaH₂PO₄) buffer, at various pHs ranging from 5-9. The solvent water peak was attenuated using pulsed field gradients or by pre-saturation.

Two-dimensional NMR based ³¹P/H Heteronuclear Single Quantum Coherence (HSSQC) spectra were acquired at
25°C, on an 800 MHz Bruker Avance spectrometer. Data were collected from 0.6 mL samples of 300–350 µM \(^{15}N\) labelled sample of apo- and holo-Ag\(^{+}\)-bound SilE in 10 mM HEPES buffer, 20 mM NaF, 10% D\(_2\)O buffer at pH 7.5. Due to the higher protein concentration required for NMR experiments, Ag\(^{+}\) was dialysed into the holo sample at a concentration a final concentration of 500 mM and excess ions were removed through further dialysis using NMR data was processed and analyzed using Topspin package (Bruker).

**Bioinformatics**

Both native protein sequences (excluding their periplasm exporting leader sequence, residues 1-20) of PcoE and SilE were aligned, using a general multipurpose alignment program for protein primary sequences – Clustal Omega, which finds the best alignment across the entire length of each sequence submitted (McWilliam et al., 2013). Their percentage identity was then calculated in Jalview (Waterhouse et al., 2009). The sequences were coloured using the “Percentage Identity” colour-scheme in Jalview too, to clarify the sequence similarities at a more obvious intensity.

Based on the SilE sequence, Jpred (a secondary structure prediction server that incorporates the Jnet algorithm to make more accurate predictions) was used to predict α-helices, β-strands and random coil (Cole et al., 2008). Additionally, the intrinsic protein disorder predictor DisEMBL, which utilises the PDB (Protein Data Bank), was used to predict disordered loops in SilE (Linding et al., 2003).

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**NOTE ADDED IN PROOF:**

Since submission of this manuscript a paper on evolution of pco/sil has been published (Staehlin et al., 2016)

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**Supporting information**

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