Novel gramicidin formulations in cationic lipid as broad-spectrum microbicidal agents

Abstract: Dioctadecyldimethylammonium bromide (DODAB) is an antimicrobial lipid that can be dispersed as large closed bilayers (LV) or bilayer disks (BF). Gramicidin (Gr) is an antimicrobial peptide assembling as channels in membranes and increasing their permeability towards cations. In mammalian cells, DODAB and Gr have the drawbacks of Gram-positive resistance and high toxicity, respectively. In this study, DODAB bilayers incorporating Gr showed good antimicrobial activity and low toxicity. Techniques employed were spectroscopy, photon correlation spectroscopy for sizing and evaluation of the surface potential at the shear plane, turbidimetric detection of dissipation of osmotic gradients in LV/Gr, determination of bacterial cell lysis, and counting of colony-forming units. There was quantitative incorporation of Gr and development of functional channels in LV. Gr increased the bilayer charge density in LV but did not affect the BF charge density, with localization of Gr at the BF borders. DODAB/Gr formulations substantially reduce Gr toxicity against eukaryotic cells and advantageously broaden the antimicrobial activity spectrum, effectively killing Escherichia coli and Staphylococcus aureus bacteria with occurrence of cell lysis.

Keywords: antimicrobial peptides, cationic bilayers, bilayer fragments, dioctadecyldimethylammonium bromide, gramicidin A, self-assembly

Introduction

Peptides used in therapy should be resistant to the action of proteases.\textsuperscript{1} Many peptides have been used in a number of biotechnological applications, ranging from neutralizing toxins to antitumor or antimicrobial agents.\textsuperscript{2} Antimicrobial peptides (AMPs) are considered an important class of molecules requiring development of suitable antibacterial formulations. AMPs have been known for many years, yet very few have been used extensively in the clinic due to their considerable toxicity and high manufacturing costs.\textsuperscript{2} Further, the development of resistance by pathogenic bacteria requires multiple lines of urgent action, such as the design and synthesis of model peptides with potent antimicrobial activity and low toxicity\textsuperscript{3} and the invention of novel supramolecular assemblies for the available peptides, allowing combination of different mechanisms of action.\textsuperscript{4–6} For example, small cationic peptides were recently shown to delocalize peripheral membrane proteins, thereby impacting important cellular processes such as respiration and cell wall biosynthesis.\textsuperscript{7} Lipid bilayer disks\textsuperscript{8} have been used as carriers for AMPs, leading to novel, nontoxic, and efficacious formulations which fully protect the peptide against degradation but preserve its activity against bacteria.\textsuperscript{9} Some polysaccharides have also been evaluated as suitable vehicles for AMPs, with confirmation of the advantages of prolonged retention of peptides at the site of application due to bioadhesion and slowing of proteolytic degradation.\textsuperscript{10,11}

Gramicidin A (Gr) is an antimicrobial peptide with a unique mechanism of hampering the function of bacterial plasma membranes, thereby preventing development...
of bacterial resistance. Gr is a 15-residue linear peptide extracted from *Bacillus brevis*, that acquires a β-helix secondary structure with an internal pore and forms transmembrane channels consisting of two Gr molecules associated head-to-head.12,13 This channel allows permeation of cations across the membrane, changing the ionic balance14 and ultimately generating the observed antibiotic activity of Gr.15,16 However, Gr is insoluble in water and highly toxic to mammalian cells17 over a range of bactericidal concentrations.18 Further, Gr was reported to be ineffective against Gram-negative bacteria.19,20 On the other hand, some cationic lipids assembling as artificial membranes in aqueous solution have been described as potent microbicidal agents against Gram-negative bacteria, but are much less efficient against Gram-positive bacteria.21–23 and fungi.24–26

In the present work, a broad-spectrum combination of Gr and dioctadecyldimethylammonium bromide (DODAB) as closed bilayers (LV) or bilayer disks (BF)27 was characterized with regard to its physical properties, antibacterial activity, and toxicity in a eukaryotic cell model, ie, *Saccharomyces cerevisiae*.28 Such novel combinations have the advantages of low toxicity and broad-spectrum activity via a mechanism that does involve bacterial cell lysis.

**Materials and methods**

**Reagents**

DODAB, gramicidin D (a peptide mixture consisting mostly of Gr), ethanol, chloroform, and 2,2,2-trifluoroethanol (TFE) were purchased from Sigma-Aldrich (St Louis, MO, USA).

**Preparation of lipid dispersions**

LV were obtained from hydration and vortexing of DODAB films with 1 mM sodium chloride aqueous solution (60°C) until the dispersions became homogeneous at a final DODAB concentration equal to 0.002 M.29,30

The ultrasonic dispersion of LV with a macrotip (85 W nominal output for 20 minutes at 70°C) followed by centrifugation (10,000 g for 60 minutes at 15°C) to eliminate titanium particles has been previously described as a procedure for yielding a BF dispersion.29 Disruption of LV in this manner generates the BF. DODAB quantitative analysis was performed by bromide microtitration, as described elsewhere.31

**Preparation and characterization of DODAB/Gr assemblies**

Aliquots of a stock solution of Gr (6.4 mM) in TFE were added to the DODAB bilayers before incubation (one hour at 70°C) at a DODAB to Gr molar ratio of 10:1. The size distribution, hydrodynamic diameter, polydispersity index, and zeta potential of the dispersions were obtained by photon correlation spectroscopy or dynamic light scattering at 90° using the Brookhaven Zeta Plus-Zeta apparatus (Brookhaven Instruments Corporation, Holtsville, NY, USA). The log-normal fitting function of the apparatus software was used to calculate the mean diameters of the size distributions.32 The zeta potential was given as $\zeta = \mu \eta / \varepsilon$, where $\mu$, $\eta$, and $\varepsilon$ are the electrophoretic mobility in 1 mM NaCl solution (25°C), the medium viscosity, and the dielectric constant, respectively.

**Evaluation of Gr incorporation in DODAB LV or BF from fluorescence spectra**

LV/Gr and BF/Gr (0.2 mM DODAB) were filtered using polycarbonate membranes (with a cutoff of 0.2 μm). The total area under the Gr fluorescence spectra before ($A_{\text{before}}$) and after ($A_{\text{after}}$) filtering gave $\%\text{Gr} = 100 \times A_{\text{after}}/A_{\text{before}}$. The fluorescence emission spectra for Gr were determined at 25°C using a F4500 fluorescence spectrofluorometer (Hitachi, Tokyo, Japan) at $\lambda_{\text{exc}} = 280$ nm. Excitation and emission slits were fixed at 2.5 nm. The molar extinction coefficient for Gr at 280 nm is 20,700 m$^3$·cm$^{-1}$.33

**Circular dichroism spectra for Gr in different types of medium**

Spectra were acquired at 25°C using a 720 spectropolarimeter (Jasco Inc, Tokyo, Japan) in a 0.1 cm quartz cell with 0.5 nm wavelength increments and a 4-second response in the 200–280 nm range (100 nm per minute). Each spectrum is the average of five scans, with a full-scale sensitivity of 10 m deg. All spectra were corrected for background by subtraction of appropriate blanks in the absence of Gr (DODAB dispersion or Gr solvent). Spectra smoothing kept the overall spectral shape. The ellipticities $\theta$ (in deg dmol$^{-1}$ cm$^{-1}$) were plotted as a function of wavelength.

**Evaluation of Gr-induced changes in permeability of DODAB closed bilayers at 2 mM DODAB**

Absorbance derived from light scattered by the LV at 400 nm was recorded as a function of time upon addition of hyperosmotic KCl or glucose solutions to LV or LV/Gr dispersions by means of a U2000 spectrophotometer (Hitachi) at room
temperature. No hyperosmotic solution was added to the reference cuvette; instead, water was added. Absorbances were normalized to the initial value. The turbidity (400 nm) kinetics was followed after adding 20 mM KCl or 40 mM D-glucose to LV in water.

**Microbial cultures**

*Escherichia coli* (ATCC 25922; American Type Culture Collection, Manassas, VA, USA) and *Staphylococcus aureus* (ATCC 25923) were incubated for 3 hours at 37°C under shaking in tryptic soy broth (Merck, Darmstadt, Germany), spread onto Mueller-Hinton agar plates (Hi-Media Laboratories Pvt, Mumbai, India), and incubated for 24 hours at 37°C. Some colonies from the plates were shaken in tryptic soy broth (160 rpm at 37°C for 2 hours) to reach the exponential phase of growth before pelleting (8,000 rpm for 15 minutes) and washing with a 0.264 M D-glucose solution. Washing was performed twice, and the bacteria were then mixed with the DODAB BF or LV with or without Gr in the same D-glucose solution. Tube 0.5 of the McFarland scale at 625 nm (about 1.5×10^8 colony-forming units [CFU]/mL) was used as the reference for preparation of the bacteria suspension. Thereafter, bacteria and the DODAB, Gr, or DODAB/Gr dispersions were allowed to interact for counting of CFU, as described in the next section.

*S. cerevisae* was cultured in yeast extract peptone dextrose medium for both broth and solid agar cultures as previously described. Some isolated colonies from a fresh *S. cerevisae* culture on solid medium were transferred to yeast extract peptone dextrose broth, and the yeast suspension was then incubated at 32°C under shaking (120 rpm for 5 hours) before centrifuging (10,000 rpm for 10 minutes), washing the pellet three times with 0.264 M D-glucose solution, and adjusting the cell concentration to 3–4×10^8 cells/mL as determined by CFU counting.

**Determination of effects of DODAB or DODAB/Gr on cell viability**

Microbes and dispersions were mixed and interacted for one hour before being diluted up to 20,000-fold for plating of 0.1 mL of each in triplicate and incubation (24 hours for 37°C) for CFU counting. Cell survival (%), taken as the mean ± standard deviation, was plotted against DODAB and/or Gr concentration. As a control for cell viability in the absence of DODAB or DODAB/Gr dispersions, a standard bacterial suspension was added to 0.264 M D-glucose solution, diluted, and spread on the agar plate.

*S. cerevisae* and the dispersions (Gr, BF, LV, BF/Gr or LV/Gr) interacted for one hour. The control sample in the absence of the antimicrobial agent was a mixture of the standard *S. cerevisae* suspension and 0.264 M D-glucose solution. Aliquots of each diluted mixture (dilution up to 1,000 fold) were plated onto yeast extract peptone dextrose agar. CFU counting was performed after incubation for 48 hours at 32°C. Cell viability (%) was taken as the mean ± standard deviation and plotted against DODAB and/or Gr concentration. The minimum bactericidal concentration (MBC) is the concentration resulting in 99% cell death.

**Determination of leakage of phosphorylated compounds from bacteria**

Bacteria were grown on plates of Mueller–Hinton agar and incubated for 18–24 hours at 37°C to prepare a cell suspension in 1 mM NaCl solution with turbidity at 625 nm adjusted to 0.400. Aliquots were then transferred to Eppendorf tubes and pelleted (6,000 rpm for 15 minutes) before resuspension in the DODAB, Gr, or DODAB/Gr dispersions in 1 mM NaCl. Bacteria and the assemblies were allowed to interact for one hour, centrifuged (6,000 rpm for 15 minutes), and used to determine the inorganic phosphorus concentration in the supernatant (ie, Pi supernatant), as previously described. As a positive (Pi) control, inorganic phosphorus was determined for bacteria in NaCl 1 mM. As a negative control, inorganic phosphorus was determined in the supernatant of bacterial suspensions after centrifugation. Final concentrations in the interaction mixtures ranged from 3.7×10^8 to 4.8×10^8 CFU/mL (*E. coli*) and from 5.2×10^9 to 5.1×10^9 CFU/mL (*S. aureus*). Bacterial lysis is related to the concentration of phosphorylated compounds in the supernatant as percent leakage =100×Pi_{supernatant}/(Pi_{control}).

**Evaluation of cell morphology by scanning electron microscopy**

Bacteria interacted with DODAB/Gr dispersions above the MBC (37°C for 2 hours) before being pelleted (12,000 rpm for 10 minutes), progressively dehydrated by ethanol as previously described, and then placed on a silicon wafer for evaporation of ethanol at room temperature, gold sputtering, and imaging by a JSM-6460LV scanning electron microscope (JEOL Ltd, Tokyo, Japan) at 20 kV.
Results
Sizing, zeta potential, polydispersity index, and gramicidin conformation and functionality in DODAB/Gr assemblies
Figure 1 shows the steps involved in preparing the DODAB/Gr dispersions. Preparation of BF/Gr and LV/Gr with insertion of Gr in LV and adsorption of Gr at the borders of BF are schematically illustrated. The open nature of the BF in the absence of Gr has been often reported in the literature, both by electron microscopy and by cryotransmission electron microscopy. The sonication procedure not only disperses the cationic lipid to promote its self-assembly as bilayer vesicles but also disrupts the closed vesicles, producing the bilayer fragments or disks.

Incorporation of the peptide in BF caused an increase in hydrodynamic diameter, whilst for LV, incorporation of Gr decreased the hydrodynamic diameter (Table 1). Gr possibly induced a slight aggregation of BF due to a small degree of intertwining between adsorbed Gr molecules at different BF. On the other hand, insertion of Gr in the LV improved the colloidal stability of the LV, possibly due to the presence of Gr tryptophan residues at the bilayer/water interface. Incorporation of Gr in DODAB BF slightly affected the zeta potential, which changed from 55±3 to 43±4 mV (Table 1). For DODAB LV, incorporation of Gr substantially increased the zeta potential from 46±2 mV to 61±3 mV. When Gr becomes incorporated in the LV bilayer, the charge density at the level of the polar heads increases for the LV/Gr dispersion but this does not occur for the BF upon incorporation of Gr: in the latter case, the charge density and zeta potential remain practically unchanged. For BF/Gr, Gr induced some aggregation of BF, as suggested by the increase in hydrodynamic diameter from 61±0 nm to 104±1 nm. This would be consistent with interactions between Gr molecules at the border of different BF causing a certain extent of aggregation. The polydispersity of the assemblies was reduced for both BF/Gr and LV/Gr as compared with the nonloaded bilayers, suggesting some conformational changes in the Gr molecules.
degree of stabilization of the DODAB dispersions induced by the peptide. For LV/Gr, the colloidal stabilization induced by the peptide is confirmed further by a reduction in hydrodynamic diameter from 379±10 nm to 247±2 nm (Table 1), indicating reduced aggregation for LV assemblies with Gr.

The intrinsic fluorescence of Gr due to tryptophans was used to determine incorporation of the peptide in the DODAB BF or LV dispersions. The filtration procedure for the BF/Gr dispersion yielded in the filtrate a mean molar percentage of 38%±11% for DODAB and 28%±1% for Gr. Within the limits of the experimental error, these two values are practically equal. The mean percentage of DODAB and Gr in the filtrate of the LV/Gr dispersion was 0%, indicating retention of LV or LV/Gr by the filter (Table 2). The 200 nm cutoff filter retained the majority of the LV/Gr but permitted flow of smaller BF/Gr assemblies. The results shown in Table 2 also demonstrate quantitative incorporation of Gr in the DODAB bilayers. When DODAB from BF is present in the filtrate, so is Gr. When DODAB from LV is retained by the filter, so is Gr, and in the same proportion (Table 2).

The fluorescent tryptophans in the Gr molecular structure may act as intrinsic labels for the peptide. Figure 2A shows the fluorescence spectra for Gr in different types of medium, such as TFE, ethanol, and DODAB dispersions. Although the four spectra share similar shapes and peaks, the intensity of fluorescence for Gr in BF is more similar to that for Gr in ethanol than for Gr in TFE (Figure 2A). For DODAB LV/Gr, tryptophan residues sense a microenvironment similar to the one in TFE. Thus, Gr perceives a more polar medium in BF than in LV.

Figure 2B shows the circular dichroism (CD) spectra of Gr in ethanol, TFE, BF, and LV. The Gr spectrum in LV has a general shape that resembles the spectrum of Gr in TFE. In BF, Gr displays a CD spectrum with the negative peak at about 230 nm, which is similar to that for Gr in ethanol (Figure 2B). These data support the conclusion that the Gr β-helix in BF senses a microenvironment more like that in ethanol than that in TFE, whereas in LV, the Gr β-helix perceives a medium more like that in TFE than that in ethanol. In the literature, the negative peak around 230 nm and the positive peak around 197 nm in ethanol typically show intertwining of Gr molecules. The CD spectrum for Gr in BF is more similar in shape to that of Gr in ethanol than that of Gr in TFE. However, in the case of Gr in BF, one should not expect to see the same extensive intertwining of Gr molecules in ethanol since the intertwining event may take place at a lower frequency at the BF borders than it does in ethanol. For example, the intertwining between Gr molecules adsorbed on different BF may be the reason for aggregation of BF with the observed increase in size (Table 1). However, this increase in size is not so large as to be consistent with extensive aggregation of BF.

The effect of adding hypertonic solutions to LV is shown in Figure 3. Turbidity at 400 nm increased with time when the water from the inner vesicle compartment flowed in accordance with the solute gradient, resulting in shrinkage of

### Table 1 Physical characterization of DODAB BF or LV, DODAB BF/Gr, or DODAB LV/Gr dispersions prepared in a 1 mM solution of NaCl

| Dispersion   | Dz (nm) | ζ (mV) | PI     |
|--------------|---------|--------|--------|
| DODAB BF     | 61±0    | 55±3   | 0.240±0.006 |
| DODAB BF/Gr  | 104±1   | 43±4   | 0.197±0.007 |
| DODAB LV     | 379±10  | 46±2   | 0.238±0.012 |
| DODAB LV/Gr  | 247±2   | 61±3   | 0.182±0.016 |

**Note:** Mean Dz, ζ, and PI are given at final DODAB and/or Gr concentrations of 2 mM and 0.2 mM, respectively.

**Abbreviations:** DODAB, dioctadecyldimethylammonium bromide; LV, closed bilayers; BF, bilayer disks; Gr, Gramicidin; PI, polydispersity index; Dz, hydrodynamic diameter; ζ, surface potential at the shear plane.

### Table 2 DODAB or Gr concentrations (%) in BF/Gr or LV/Gr dispersions before and after filtering

|          | BF/Gr | LV/Gr |
|----------|-------|-------|
|          | Original dispersion | Filtrate | Original dispersion | Filtrate |
| DODAB/mM | 0.24±0.08 | 0.09±0.01 | 0.21±0.07 | 0 |
| Integral of Gr fluorescence spectrum | 4133±23 | 1148±14 | 3382±21 | 0 |
| % DODAB in filtrate | 38±1 | 0 |
| % Gr in filtrate | 28±1 | 0 |

**Abbreviations:** DODAB, dioctadecyldimethylammonium bromide; LV, closed bilayers; BF, bilayer disks; Gr, Gramicidin.
LV (Figure 3). It has been previously reported for vesicles of similar size that turbidimetry allows monitoring of vesicle shrinkage or swelling, which corresponds respectively to an increase or decrease in turbidity kinetics at 400 nm.29,39 In fact, the osmotic behavior of LV depended on the osmolarity of the outer medium, but in the presence of Gr, this behavior changed significantly, ie, instead of shrinking, the LV swelled on addition of the hypertonic external medium (Figure 3).

The Gr channel conformation in LV modifies the usually low permeation of cations through the DODAB bilayer. Permeation of D-glucose through the DODAB LV/Gr bilayer also increases substantially in comparison with permeation through the DODAB LV bilayer. The entry of D-glucose or KCl solutes via the Gr channel from the outer to the inner vesicle compartment is accompanied by entry of water, which causes the vesicle to swell (Figure 3).

### Antimicrobial activity, mechanism of action, and differential cytotoxicity of DODAB/Gr assemblies

The antimicrobial activity of DODAB, Gr, and DODAB/Gr assemblies was evaluated against *E. coli*, *S. aureus*, and *S. cerevisiae* in order to gain further insights into the differential cytotoxicity of the combined formulations (Figure 4). There was poor Gr activity against *E. coli* and *S. cerevisiae* as shown from the survival of microbes (%) over a range of Gr concentrations, in contrast with the action of Gr on *S. aureus*, consistent with previous reports in the literature.18–20

The activity of DODAB as shown in Figure 4 was also consistent with previous data in the literature reporting its high activity against *E. coli*21,22 and poor activity against the yeast.24–26 The novelty of these combinations was the broadening of antimicrobial activity to encompass *E. coli* and *S. aureus* as representatives of Gram-negative and Gram-positive bacteria. Of note, the microbicidal activity occurred over a range of low Gr and DODAB concentrations which are not toxic to *S. cerevisiae* (Figure 4).

The MBCs taken from the viability curves for all dispersions are shown in Table 3. The MBCs are lower for DODAB and Gr in combination in comparison with those obtained for the separate DODAB and Gr dispersions (Table 3). The lowest MBCs were obtained for DODAB BF/Gr, which revealed superior performance for delivering Gr to bacteria in comparison with DODAB LV/Gr. The high DODAB concentrations needed to kill *S. aureus* reconfirm the emergence

![Figure 3](image-url)
Table 3 Minimum bactericidal concentrations (in μM) and cell lysis at the minimum bactericidal concentration (% L) for DODAB, Gr, and DODAB/Gr dispersions

| Dispersion | DODAB E. coli | Gr E. coli | DODAB S. aureus | Gr S. aureus | % L E. coli | % L S. aureus |
|------------|---------------|------------|-----------------|--------------|-------------|--------------|
| DODAB BF   | 5.5           | –          | >200            | –            | 2.5         | –            |
| DODAB LV   | 20            | –          | 34.0            | –            | 1.0         | 0            |
| Gr         | –             | >20        | –               | 4.0          | 0           | 0            |
| DODAB BF/Gr| 3.2           | 0.32       | 6.8             | 0.68         | 2.5         | 0            |
| DODAB LV/Gr| 8.0           | 0.80       | 32.0            | 3.2          | 0           | 0            |

Abbreviations: DODAB, dioctadecyldimethylammonium bromide; LV, closed bilayers; BF, bilayer disks; Gr, Gramicidin; E. coli, Escherichia coli; S. aureus, Staphylococcus aureus.
of resistance of Gram-positive bacteria to cationic antimicrobial agents. In this respect, delivering Gr becomes very important since Gr does not bear cationic moieties and acts by a different mechanism. The excellent activity of DODAB against Gram-negative bacteria complements the poor activity of the peptide against these bacteria. The excellent activity of Gr against Gram-positive bacteria complements the poor activity of DODAB against these bacteria. The combined formulations barely affected the S. cerevisae cells, suggesting limited toxicity if used for further therapeutic application.

In order to shed some light on the mechanism of antimicrobial action for the assemblies, leakage of phosphorylated compounds from bacteria was determined over a range of DODAB concentrations (Figure 5), and no cell rupture was seen over a range of low DODAB concentrations. This remained so up to the MBC, as also seen in Table 3. Above the MBC, the leakage increased, suggesting some important lytic events taking place in the cells. The bacterial morphology examined by scanning electron microscopy before and after interaction with DODAB/Gr at DODAB and Gr concentrations above the MBCs also suggested some cellular distortions departing from the morphology of untreated cells (Figure 6).

### Discussion

The physical behavior of Gr in the DODAB dispersions was very similar to that previously reported for Gr in composite bilayers of phospholipid and DODAB at a molar proportion of 1:1. Fluorescence and CD data reported the Gr microenvironment. For DODAB BF, the Gr helical moiety perceives the hydrophobic borders of the bilayer disks and the aqueous medium surrounding the disks. The intertwined Gr conformation in BF was depicted in the CD spectra. The Gr beta-helix in LV or TFE was demonstrated by the CD positive peak around 225 nm and associated with the functional channel present in LV (Figure 2B). Tryptophans at the membrane/water surface help to ensure appropriate conformation and activity in the Gr channel. Gr fluorescence is similar in LV and TFE, showing localization of the tryptophans in a hydrophobic medium (Figure 2A). In contrast, Gr fluorescence is also similar in BF and ethanol, showing localization of the tryptophans in a more polar medium (Figure 2A). Gr possibly attaches to the edges of the disks at low concentrations, but with increasing Gr concentration, other Gr molecules from the outer solution can intertwine with those already attached at the disk border.

For LV, shrinking or swelling causes changes in the turbidity of the dispersions. In accordance with the Joebst equation, the turbidity of the spheres in the dispersion varies with \(1/R^2\), where \(R\) is the mean particle radius and is useful for inferring changes in vesicle size taking place upon establishment of osmotic gradients through the LV bilayers. DODAB LV shrinkage and swelling would correspond to the kinetics of increasing and decreasing turbidity, respectively (Figure 3). The Gr inserted in LV form channels, that change the LV permeation to solutes, such as D-glucose and KCl consistent with previous reports of similar vesicular dispersions in the presence of Gr.

Hydrophobic drugs solubilize at the hydrophobic borders of BF, in contrast with the absence of drug solubilization in LV. Given the hydrophobic nature of the Gr beta-helix, the behavior of Gr in the DODAB bilayers is similar to that of hydrophobic drugs in good organic solvents and in DODAB bilayers. However, delivery of Gr to the bacterial cells seems to be favored by its combination with BF, possibly due to peripheral localization of Gr in the BF (Table 3).

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**Figure 5** Leakage of phosphorylated compounds (%) from bacteria for different dispersions at 3.7 × 10⁴ to 4.8 × 10⁹ CFU/mL (E. coli and S. aureus). The dispersions were tested over a range of DODAB or DODAB/Gr concentrations after one hour of interaction with both types of bacteria. As a control, over a range of Gr concentrations (10⁻² to 10⁻⁴ mM), no leakage was detected from bacteria after the same interaction time.

**Abbreviations:** CFU, colony-forming units; DODAB, dioctadecyldimethylammonium bromide; LV, closed bilayers; BF, bilayer disks; Gr, Gramicidin.
The antimicrobial activity of the DODAB cationic lipid observed at a very low DODAB concentration is related to the quaternary nitrogen in its molecular structure. The differential cytotoxicity of DODAB has been systematically studied, eg, at 0.5 mM, DODAB kills about 50% of fibroblasts cultured as a subconfluent monolayer. The antimicrobial activity of Gr over the micromolar range of concentrations against Gram-positive bacteria has also been well established, as well as its differential cytotoxicity. The minimum inhibitory concentration of Gr against S. aureus is 2.5 μM. The survival of mammalian HeLa cells at this concentration is 15%, indicating that the peptide is highly toxic. In addition, there is about 50% survival of cultured renal cells at 2.5 μg/mL Gr, reconfirming its high toxicity at concentrations that kill the bacteria. Here the MBC for Gr against S. aureus is about 4 μM (Figure 4, Table 3) and is consistent with the previously reported minimum inhibitory concentration of 2.5 μM. Gr is indeed very toxic in mammalian cells, and also against eukaryotic cells, as exemplified by its activity against S. cerevisiae (Figure 4). However, in formulations containing DODAB, this toxicity is substantially reduced, as indicated by comparison of the viability curves for S. cerevisiae using Gr alone and using DODAB/Gr (Figure 4). Cell viability remained very high for both DODAB BF/Gr and DODAB LV/Gr, clearly showing improvement in delivery of Gr by the DODAB/Gr combinations. Further, DODAB is not only a vehicle but also displays good activity against E. coli, thus broadening the spectrum of antimicrobial activity for Gr.

The mechanism of action of these novel formulations involves bacterial cell lysis since leakage of phosphorylated compounds took place from the MBC values (Figure 5, Table 3). Some lysis was detected at doses above the MBC, and visualization of cells by scanning electron microscopy indeed showed distortions in cell morphology (Figure 6). The detergent-like mechanism often ascribed to the interaction between antimicrobial peptides and surfactants seems to be important for determining antimicrobial activity. For the DODAB/Gr combinations, the mechanism involved affects membrane function and selectivity in the transport of ions and nutrients and ion distribution in the cell. Death of bacterial cells possibly takes place with some substantial membrane rupture and leak of intracellular bacterial compounds. Eventually, damage to certain transmembrane proteins may also be involved in the death mechanism.

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Disclosure
The authors report no conflicts of interest in this work.

References
1. Bracci L, Falciani C, Lelli B, et al. Synthetic peptides in the form of dendrimers become resistant to protease activity. J Biol Chem 2003; 278(47):46590–46595.
2. Roscia G, Falciani C, Bracci L, Pini A. The development of antimicrobial peptides as new antibacterial drugs. Curr Protein Pept Sci. 2013; 14(8):641–649.
3. Kim S-J, Kim J-S, Lee Y-S, et al. Structural characterization of de novo designed LS5K5W model peptide isomers with potent antimicrobial and varied hemolytic activities. *Molecules*. 2013;18(18):859–876.

4. Carmona-Ribeiro AM. Biomimetic nanoparticles: preparation, characterization and biomedical applications. *Int J Nanomedicine*. 2010;5:249–259.

5. Carmona-Ribeiro AM, Barbassa L, Melo LD. Antimicrobial biomimetics. In: Cavrak M, editor. Biomimetic Based Applications. Rijeka, Croatia: InTech; 2011.

6. Carmona-Ribeiro AM, de Melo Carrasco LD. Cationic antimicrobial polymers and their assemblies. *Int J Mol Sci*. 2013;14(5):9906–9946.

7. Wenzel M, Chiriac AI, Otto A, et al. Small cationic antimicrobial peptides delocalize peripheral membrane proteins. *Proc Natl Acad Sci U S A*. 2014;111(14):E1409–E1418.

8. Carmona-Ribeiro AM. Lipid bilayer fragments and disks in drug delivery. *Curr Med Chem*. 2006;13(12):1359–1370.

9. Zetterberg MM, Reijmar K, Praenting M, Engstroem A, Andersson DI, et al. Small cationic antimicrobial peptides: a potential approach to the treatment of oral infections. *Dent Microbiol*. 2013;95:5361–5366.

10. Ruissen AL, van der Reijden WA, van’t Hof W, Veerman EC, Nieuwenhuis HM, et al. Evaluation of the use of xanthan as vehicle for cationic antimicrobial peptides. *J Control Release*. 2011;156(3):323–328.

11. Ruissen AL, van der Reijden WA, van’t Hof W, Veerman EC, Nieuwenhuis HM, et al. Evaluation of the use of xanthan as vehicle for cationic antimicrobial peptides. *J Control Release*. 1999;60(1):49–56.

12. Gaweide PV, Leung KP, Madhavatha S. Antimicrobial and anti-lingual activity of Dispersin B-α-KSL-W peptide-based wound gel against chronic wound infection associated bacteria. *Curr Microbiol*. 2014;68(5):635–641.

13. Urban BW, Hladky SB, Haydon DA. Ion movements in gramicidin pores. *Biochemistry*. 1991;30(13):2713–2720.

14. Clement NR, Gould JM. Kinetics for development of gramicidin-induced ion permeability in unilamellar phospholipid vesicles. *Biochim Biophys Acta*. 2011;1806(3):331–354.

15. Clement NR, Gould JM. Kinetics for development of gramicidin-induced ion permeability in unilamellar phospholipid vesicles. *Biochim Biophys Acta*. 2011;1806(3):331–354.

16. Hamada T, Matsunaga S, Fujiwara M, et al. Solution structure of polyethylene glycol A, a highly cytotoxic nontoxic polypeptide from marine sponge. *J Am Chem Soc*. 2010;132(37):12941–12945.

17. Sorochkina AI, Plotnivok EY, Rokitskaya TI, et al. N-terminally modified highly cytotoxic amphiphilic antimicrobial peptide based on gramicidin A as potential systemic antibiotics. *Antimicrob Agents Chemother*. 1999;33:249–256.

18. Wang F, Qin L, Pace CJ, Wong P, Malonis R, Gao J. Solubilized gramicidin A as potential systemic antibiotics. *Antimicrob Agents Chemother*. 1999;33:249–256.

19. Rammelkamp CH, Weinstein L. Toxic effects of tyrothricin, gramicidin A, and related compounds. *J Infect Dis*. 1942;71(2):166–173.

20. Martius LMS, Mamizuka EM, Carmona-Ribeiro AM. Cationic vesicles as bactericides. *Langmuir*. 1997;13(21):5583–5587.

21. Carmona-Ribeiro AM, Barbassa L, Melo LD. Antimicrobial biomimetics. In: Cavrak M, editor. Biomimetic Based Applications. Rijeka, Croatia: InTech; 2011.

22. Campanhã MT, Mamizuka EM, Carmona-Ribeiro AM. Interactions between cationic liposomes and bacteria: the physical-chemistry of the bactericidal action. *J Lipid Res*. 1999;40(8):1495–1500.

23. Melo LD, Mamizuka EM, Carmona-Ribeiro AM. Antimicrobial particles from cationic lipid and polyelectrolytes. *Langmuir*. 2010;26(14):12300–12306.

24. Campanhã MT, Mamizuka EM, Carmona-Ribeiro AM. Interactions between cationic vesicles and Candida albicans. *J Phys Chem B*. 2001;105(34):8230–8236.

25. Melo LD, Carmona-Ribeiro AM. Fungalicidal nanoparticles of low toxicity from cationic lipid and polyelectrolytes. *NIST Nanotech*. 2012;3:350–353.

26. Carmona-Ribeiro AM, Carrasco LDM. Fungalicidal assemblies and their mode of action. *OA Biotechnology*. 2013;2(3):25.

27. Carmona-Ribeiro AM. Synthetic amphiphile vesicles. *Chem Soc Rev*. 1999;21:209–214.

28. Buschini A, Poli P, Rossi C. *Saccharomyces cerevisiae* as an eukaryotic cell model to asses cytotoxicity and genotoxicity of three anticancer anthraquinones. *Mutagenesis*. 2003;18(1):25–26.

29. Carmona-Ribeiro AM, Chaimovich H. Preparation and characterization of large dioctadecylmethylammonium chloride liposomes and comparison with small sonicated vesicles. *Biochim Biophys Acta*. 1983;733(1):172–179.

30. Sobral CN, Soto MA, Carmona-Ribeiro AM. Characterization of DODAB/DPPC vesicles. *Chem Phys Lipids*. 2008;152(1):38–44.

31. Schales O, Schales S. A simple and accurate method for the determination of cholesterol in biological fluids. *J Biol Chem*. 1941;140:879–884.

32. Grabowski E, Morrison I. Measurements of suspended particles by quasi-elastic light scattering. *J Phys Chem B*. 2004;108(41):15795–15807.

33. Killian JA, Prasad KU, Hains D, Urry DW. The membrane as an environment of minimal interconversion. A circular dichroism study on the solvent dependence of the conformational behavior of gramicidin in didodecylphosphatidylcholine membrane bilayers. *Biochemistry*. 1988;27(13):4848–4855.

34. Carmona-Ribeiro AM. Preparation and characterization of biomimetic nanoparticles for drug delivery. *Methods Mol Biol*. 2012;906:283–294.

35. Hoque J, Akkapeddi P, Yalagadda V, Uppu DS, Kumar P, Haldar J. Cleavable cationic antibacterial amphiphiles: synthesis, mechanism of action, and cytotoxicities. *Langmuir*. 2012;28(33):12225–12234.

36. Carmona-Ribeiro AM, Castuma CE, Sesso A, Schreier B. Bilayer structure and stability in dilaureyl phosphate dispersions. *J Phys Chem B*. 1997;101:5361–5366.

37. Andersons M, Hammarstrom L, Edwards K. Effect of bilayer phase transitions of vesicle structure, and its influence on the kinetics of viologen reduction. *J Phys Chem*. 1995;99(39):14531–14538.

38. Abdul-Manant N, Hinton JF. Conformation states of gramicidin A along the “accidental” pathogen. *Nat Rev Microbiol*. 2009;7(8):555–567.