Formulation and Evaluation of Microsponges Gel of Havan Ash for the Treatment of Acne

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ABSTRACT

The aim of this study was to develop the Microsponges containing Havan ash composed gel formulation for the treatment of Acne. Therefore, the topical formulation containing microsponges of Havan Ash will be formulated and evaluated. The preliminary investigation was carried out for the formulation of Havan ash loaded Microsponges by using quasi emulsion solvent diffusion method (MSF1-MSF6). In the preformulation studies of Havan ash the physical description and organoleptic properties, pH, acid insoluble ash, water-soluble ash, IR spectroscopy, identification test, rheological study, atomic absorption spectroscopy is also carried out. On the basis of particle size analysis of Microsponges, percentage yield formulation MSF5 containing Microsponges formula was selected for composition of topical gel formulation. Thus the different gel base formulation (G1-G3) using Carbopol-934 (1,1.5,2.0%) was prepared by emulsification method. By considering all the relevant, physicochemical parameters, G2 gel base was selected for further loading of Havan ash containing Microsponges. The MSF5 formulation was loaded into the selected gel base G2 (1.5%). Then the formulation and evaluation of Havan ash microsponges loaded gel was done. The formulation F3 has better results than other 4 formulations. F3 have its appearance silver colour, consistency very good, Grittiness 8, homogeneity good, wash ability very good, pH 6.3, Spreadabilty (g.cm/sec) 14.4 ± 0.77 7 and viscosity (cps) 18251 ± 50.12, have good result of psychometric analysis. With the revealed results by different evaluation parameters, it is concluded that microsponges drug delivery system has become highly competitive and rapidly evolving technology and more research are carrying out to optimize cost-effectiveness and efficacy of the therapy.

Keywords: Havan ash, Antimicrobial, Microsponges, Acne vulgaris, Topical gel.

INTRODUCTION

According to Hindu mythology Bhasma or vibhuti is the sacred ash from the dhuni or fire where special wood, ghee, herbs, grains and other auspicious and purifying items are offered for worship along with mantras. Culturally, we attach a lot of importance to the holy ash on the forehead. Applying vibhuti or bhasma or the holy ash is a common practice in India. In older times RishiMuni, ancient scholars and physicians used to recommend Hawan for mental peace and well-being. Gayatri Mantra also tells that sughanahim (aroma, fragrance) puishthivardhaman (gives rise to good health). Om triambkum yajamahe, sughandhim

Hawan is an ancient ritual which is one of the most important Vedic ritualistic sacrifices which involves lighting fire in a rectangular Homa Kunda. Hawan is an ancient ritual which is performed to purify the atmosphere and the environment. Hawan is also believed to help plants to acquire a protective coating against pests and diseases. Hawan Samagri is a collection of...
plant material for burning in a Hawan along with other herbal embodiments or sacrificial fire. These herbs have antibacterial, antiviral, and many other therapeutic properties. When the Hawan Samagri burn, their smoke clears toxins and harmful micro-organisms from the environment. Since long back in our traditional treatise such Hawan ash had been used for various dermal disorder like, wound healing, eczema, scabies, as moisturising agent and skin protective antiacline, anti-wrinkle, sun protective agent etc. Hawan Ash have some medicinal properties like antibacterial, antiviral, antifungal, antimicrobial, anti-itching etc. It has antibacterial activity which used as an active ingredient for the treatment of Acne Vulgaris.  

From that traditionally method we get idea to formulate topical pharmaceutical dosages form containing hawan ash (vibhuti) for prevention of skin disorder.

During the past few years, interest in the development of Novel Drug Delivery Systems for existing drug molecules has been renewed. The development of a Novel Delivery System for existing drug molecules not only improves the drug’s performance in terms of efficacy and safety but also improves patient compliance and overall therapeutic benefit to a significant extent.

When properly designed and developed for a particular drug, novel drug delivery system can overcome specific hurdles associated with conventional methods of delivery, e.g. drugs that undergo partial or complete degradation before reaching the site of action could be effectively delivered with improved bioavailability by using the novel concepts of timed, pulsatile, or targeted release.

Time is the fastest runner and as we are running in the 21st century, an era where the technology and research has enriched the quality of our life with the advancement in the pharmaceutical and medical science. This dedication has sorted out many health related problems. In today's life every person whether its man or woman dreamt of a seamless and effective management of the disease. Currently, research is aimed at the development of drug delivery system with maximum therapeutic benefits for safe and effective management of the disease.

A Microsponge Delivery System (MDS) is patented, highly cross-linked, porous, polymeric microspheres that can entrap wide range of actives and then release them onto the skin over a time and in response to trigger. This system was employed for the improvement of performance of topically applied drugs. It is a unique technology for the controlled release of topical agents and consists of microporous beads, typically 10-25 microns in diameter, loaded with active agent. When Microsponge delivery system applied to the skin, the release of drug can be controlled through diffusion or other variety of triggers, including rubbing, moisture, pH, friction, or ambient skin temperature.

Microsponge technology offers:
- Enhanced product performance.
- Extended release.
- Reduced irritation and hence improved
- Improved product elegance.
- Oil control: Microsponge can absorb oil up to 6 times its weight without drying.
- Improved formulation flexibility.
- Improved thermal, physical, and chemical stability.
- Flexibility to develop novel product forms.
- Microsponge systems are non-irritating, non-mutagenic, non-allergenic and non-toxic.

Properties of the Actives for the Entrapment into Microsponges

- It should be either fully miscible in a monomer or capable of being made miscible by the addition of a small amount of a water-immiscible solvent.
- It should be water immiscible or at most only slightly soluble.
- It should be inert to monomers and should not increase the viscosity of the mixture during formulation.
- It should be stable when in contact with the polymerization catalyst and under conditions of polymerization.
- The spherical structure of the microsponges should not collapse.

Acne Vulgaris

Acne Vulgaris (commonly called acne) is a follicular skin disease that mainly affects the Pilosebaceous unit of face, neck, and trunk and is characterized by inflammatory (papules, pustules, nodules, and cysts) and non-inflammatory lesions (Seborrhea and Comedones) and scarring. Acne may be of inflammatory or non-inflammatory forms. Although in many patients, acne can be limited to a couple of papules or Comedones, serious illness can lead to disfiguring scars on the face. Despite not being a life-threatening disease, acne can have severe psychosocial consequences causing low self-esteem, social isolation, and depression.

Acne, a multifarious chronic inflammatory state, is said to be happening within a pilosebaceous unit (PSU) including hair, hair follicles, sebaceous gland (SG) of the skin, characterized by non-inflammatory lesions blackheads, whiteheads and inflammatory Lesions, pustules, Pustul, nodules, and cysts. Acne vulgaris is characterized by various clinical
conditions such as scaly red skin (Seborrhea), erythematous papules and pustules, Comedones, nodules, deep pustules, and sometimes pimples [shown in figure 1]. Acne is associated mostly with Propionibacterium acnes (P. acnes) which produces inflammation via release of extracellular enzymatic products like proteases, lipases, and hyaluronidases. About 94-95% of the pubertal population, 20-40% of adults and < 25% of women suffered from acne. Even after 25 years of age, women can suffer from is termed as adult female acne. Primarily four causes are critically responsible for the growth of acne lesion.

![Different type of acne](image)

**Figure 1: Different type of acne**

The treatment of acne can be given by topical or systemic therapy. The topical therapies include antibiotics, anti-inflammatory and Comedolytic agent, chelating agent, natural product 16-20.

Traditional herbal medicine has been used since ancient time in many parts of the world where access to formal and modern healthcare is limited 21. Traditional herbal medicines provide an interesting, largely unexplored source for new drug development 22. Medicinal plants have a long history and have been shown to possess less side effects. These herbs negligible the adverse effect as compared with modern medicines and become another important aspect to treat acne vulgaris. Upcoming years herbal therapies are gaining attention of academician, industrialist, cosmetician, researches, dermatologist and scientist for treatment of acne. Acne can be cured by herbs by both externally and internally. Topical treatment of herbs is first choice of customers as it is ease for application and it supresses the bitter taste of herbal formulation. Because herbs are safe, efficacious and the added advantage of multifunctionality, herbs are increasingly being used in mainstream cosmetic products, including acne-fighting compositions 23-24.

Microsponge can be prepared with two methods i.e. one step process (liquid-liquid polymerization method) and two-step process (quasi emulsion solvent diffusion method). Most common and feasible method used for preparation of microsponges is quasi emulsion solvent diffusion method.

**METHODS**

**Method of Preparation of Microsponge**

Microsponges of Havan Ash and Eudragit RL 100 was prepared by quasi-emulsion solvent diffusion method according to the formula given in table no 1, the process involved formation of quasi-emulsion of two different phases i.e. internal phase and external phase similar to emulsions. Table no 2 gives the detailed information about the prepared formulations.

1. **Preparation of internal phase:** the phase was consisted of drug (Havan Ash), polymer (Eudragit RL 100) and solvent (ethanol and dichloromethane in ratio 1:1). To prepare this phase, Eudragit RL 100 was dissolved in the mixture of solvents and then drug was further added to it and dissolved under sonication.

2. **Preparation of external phase (aqueous phase):** for the preparation of aqueous phase, weighed quantity of polyvinyl alcohol was taken and dissolved in 50ml of water in beaker.

3. **Mixing:** The internal organic phase was poured into the external aqueous phase by drop wise.

4. **Stirring:** The stirring was continued up to 6 hrs till the insoluble, rigid microparticles i.e. microsponges is formed.

5. **Filtration:** The mixture was allowed to stir until the foam settled down and after the complete evaporation of dichloromethane the mixture was filtered with whatmann filter paper (0.45 µm).

6. **Drying:** The microsponges were then dried in an air heated oven. 27-28
Table 1: Formulation of Microsponges of Havan Ash

| Sr. No. | Ingredient (mg/ml/gm) | F1       | F2       | F3       | F4       | F5       |
|---------|----------------------|----------|----------|----------|----------|----------|
| 1.      | Havan ash:           | 1:01     | 1:2      | 1:03     | 1:04     | 1:05     | 1:06     |
|         | Eudragit RL 100      |          |          |          |          |          |          |
| 2.      | Havan ash            | 500      | 500      | 500      | 500      | 500      | 500      |
| 3.      | Eudragit RL 100      | 50       | 100      | 150      | 200      | 250      | 300      |
| 4.      | Dichloromethane:     | 1:1      | 1:1      | 1:1      | 1:1      | 1:1      | 1:1      |
|         | Ethanol (5ml)        |          |          |          |          |          |          |
| 5.      | Polyvinylalkohol (mg)| 200      | 200      | 200      | 200      | 200      | 200      |
| 6.      | Distilled water (ml) | 200      | 200      | 200      | 200      | 200      | 200      |

Optimization of Microsponges

- **Particle size analysis of Microsponges**
  The particle size of the Microsponge was determined by optical microscopy and the microsponges were found to be uniform in size. The average particle size of all formulations ranges from 27.5 µm to 43.9 µm.

- **Percentage yield**
  It is calculated to know about the efficiency of any method, thus it helps in selection of appropriate method of production. After the preparation of formulations, the Practical yield was calculated as Microsponges recovered from each preparation in relation to the sum of starting material (Theoretical yield). It can be calculated using following formula.

  \[
  \text{Percentage yield} = \frac{\text{Practical yield} \times 100}{\text{Theoretical yield (drug + polymer)}}
  \]

Formulation and Evaluation of Gel Base

The loss of product was due to the formation of some agglomerates and polymer adherence to the container as a result of a viscous nature of slurry. It is calculated to know about the efficiency of any method, thus it helps in selection of appropriate method of production and in this experiment, it is revealed that with increase in polymer ratio the percent yield also increases.

Table 2: Formulation of gel base

| Sr. No. | Ingredient (mg/ml/gm) | Code | G1 | G2 | G3 |
|---------|----------------------|------|----|----|----|
| 1.      | Carbopol 934         |      | 1% | 1.5%| 2% |
| 2.      | Glycerine (ml)       |      | 2.5| 2.5| 2.5|
| 3.      | Alcohol (ml)         |      | 2  | 2  | 2  |
| 4.      | Methyl paraben (g)   |      | 0.1| 0.1| 0.1|
| 5.      | Triethanolamine (ml) |      | 2  | 2  | 2  |
| 6.      | Distilled water      |      | q.s| q.s| q.s|
Evaluation of Gel Base

The gel bases were evaluated for physical appearance, grittiness, pH, spreadability and viscosity.

- **Physical appearance**
  Gel base was evaluated visually for color, homogeneity and smoothness.

- **Grittiness**
  A small amount of gel was taken and spread between two glass slides free from grease and was observed against diffused light to check for presence of foreign particle.

- **Determination of pH**
  The sample of gel (5g) was weighed and transferred into a 100 ml of beaker. The distilled water of 45 ml was added to the beaker and maintained to 45°C with constant stirring using glass rod for 15 minute on a heating mantle. The solution was filtered and pH was measured using digital pH meter at 27°C.

- **Spreadability**
  Spreadability of gel was determined by the apparatus which consists of a wooden block, which is attached to a pulley at one end. Spreadability was measured on the basis of ‘Slip’ and ‘Drag’ characteristics of gel. A ground glass slide was fixed on the wooden block. All the sample (about 1g) were applied in between these two glass slides and they were pressed together so as to expel the air and to provide a uniform thickness of gel by placing 100 g of weight for 5 minutes. The top glass slide has the same dimension as that of the fixed ground slide. Therefore, a weight (10gm) was added to the pan and the top glass slide was subjected to pull with the help of string attached to the hook (Fig 4). The time in which the upper glass slide moves over the lower plate to cover a distance of 10 cm is noted. The spread ability (S) can be calculated using the formula. The following equation was used for the purpose:

\[ S = \frac{M \times L}{T} \]

Where, S= spreadability  
M= weight (gram)  
L= length (cm)  
T= time (seconds/minutes)

- **Determination of viscosity**
  The gel base formulation was evaluated for viscosity (in cps) using Brookfield rheometer (Conc. and plate) (R/S Plus). The sample of 1 gram (maintained at 25 ± 1°C) was placed on the plate and conc. was rotated at 10 rpm.

Formulation and Evaluation of Havan Ash Microsponges Loaded Gel Base

| Sr. No. | Ingredient                  | F1    | F2    | F3    | F4    |
|---------|-----------------------------|-------|-------|-------|-------|
| 1.      | Havan ash Microsponges     | 550   | 650   | 750   | 850   |
| 2.      | Carbopol 934                | 1.5%  | 1.5%  | 1.5%  | 1.5%  |
| 3.      | Glycine (ml)                | 2.5   | 2.5   | 2.5   | 2.5   |
| 4.      | Alcohol (ml)                | 2     | 2     | 2     | 2     |
| 5.      | Methyl paraben (g)          | 0.1   | 0.1   | 0.1   | 0.1   |
| 6.      | Triethanolamine (ml)        | 2     | 2     | 2     | 2     |
| 7.      | Distilled water             | q.s   | q.s   | q.s   | q.s   |
Evaluation of Havan Ash Microsponges Loaded Gel

- **Determination of Organoleptic Characteristics.**

  All blank formulations (i.e., formulations without any active ingredients or preservatives) and drug-loaded formulations were tested for physical appearance, colour, texture, phase separation, and homogeneity. These characteristics were evaluated by visual observation. Homogeneity and texture were tested by pressing a small quantity of the formulated cream and gels between the thumb and index finger. The consistency of the formulations and presence of coarse particles were used to evaluate the texture and homogeneity of the formulations. Immediate skin feel (including stiffness, grittiness, and greasiness) was also evaluated.

- **Determination of pH**

  One gram of each formulation (including the blank, i.e., formulation without any active ingredients or preservatives, and drug-loaded formulation) was dispersed in 25 mL of distilled water, and the pH was determined using a pH meter.

- **Psychometric Evaluation**

  Study design

  This prospective study was conducted at the department of pharmaceutics, laureate institute of pharmacy Kathog (HP) as per the ethical guidelines. Twelve human volunteers (6 females and 6 males in the average age group of 20-24) were selected for this study by making three groups for read through the individual variation. All the subjects were tested exclusively using the prepared gel formulation coded as F1, F2, F3, F4 for the psychometric parameters according to Hedonic Scale values.

  - Extremely liking: 8-9
  - Between extremely liking and medium: 6
  - Medium: 4-5
  - Between medium and dislikes: 3
  - Dislike: 1-2

  Psychometric analysis includes: colour, odour, texture, wetness, gloss, stickiness, slipperiness, firmness, appearance, rub-out, after-feel effect and pick-up and also to observe the initial compliance and safety of the gel. An application site is having a 2 cm² sample area. The study was conducted at 25°C.

- **Wash ability**

  Formulations were applied on the skin and then ease and extent of washing with water were checked manually.

- **Determination of spreadabilty**

  Spreadability of gel was determined by the apparatus which consists of a wooden block, which is attached to a pulley at one end. Spreadability was measured on the basis of ‘Slip’ and ‘Drag’ characteristics of gel. A ground glass slide was fixed on the wooden block. All the samples (about 1g) were applied in between these two glass slides and they were pressed together so as to expel the air and to provide a uniform thickness of gel by placing 100 g of weight for 5 minutes. The top glass slide has the same dimension as that of the fixed ground slide. Therefore, a weight (10gm) was added to the pan and the top glass slide was subjected to pull with the help of string attached to the hook (Fig 8). The time in which the upper glass slide moves over the lower plate to cover a distance of 10 cm is noted. The spread ability (S) can be calculated using the formula. The following equation was used for the purpose:

  \[
  S = \frac{M \times L}{T}
  \]

  Where, S= spreadability
  M= weight (gram)
  L= length (cm)
  T= time (seconds/minutes)

- **Determination of viscosity**

  The gel base formulation was evaluated for viscosity (in cps) using Brookfield rheometer (Conc. and plate) (R/S Plus). The sample of 1 gram (maintained at 25 ± 1°C) was placed on the plate and conc. was rotated at 10 rpm.

- **Stability studies**

  To access the drug and formulation stability, the stability studies were carried out as per ICH guidelines. The gel filled in collapsible tube and kept in humidity chamber maintained at 40°C ± 2°C and 75 ± 5 % RH for a period of 6 months. Samples were withdrawn at time intervals of 0, 3 and 6 months evaluated for its physical appearance, consistency, pH, viscosity.

**RESULTS AND DISCUSSION**

**PREFORMULATION STUDIES**

**Table 4: Physical description and organoleptic properties**

| Colour | Grey |
|--------|------|
| Odour  | Good, Ambrosial aromatic |
| Smell  | Good |
| Taste  | Grey |
- **Determination of pH, Acid Insoluble Ash, Water-soluble ash**

The pH, Acid insoluble ash, Water insoluble ash is given in table no. 5

**Table 5: Shown pH, Acid Insoluble Ash, Water-soluble ash value**

| pH  | Acid Insoluble Ash | Water-soluble ash |
|-----|--------------------|--------------------|
| 9.2 | 0.15               | 0.03               |

- **IR Spectroscopy**

The IR spectroscopy of Havan ash is given in fig 6 and describe about its peaks in Table no. 6

**Table 6: Peak Table of Havan Ash**

| Peaks    | Functional group         |
|----------|--------------------------|
| 2873 (3000-2840) | C-H                      |
| 1798     | C=O                      |
| 1606     | C=H                      |
| 1460     | C-H BENDING              |
| 1410     | O-H BENDING              |
| 1287     | C-O STRECHING            |
| 1066     | S=O                      |
| 989      | C=C                      |
| 876      | C-H                      |

![Figure 6: IR of Havan ash](image)
• **Identification test** (presence of Zn, Na, K, Fe, Ca, Mg, Al etc. depends upon ingredients)

The identification of lead, Magnesium, Sulphate, Zinc, Ferrous, Aluminium, Calcium, Sodium and Potassium was performed. The result is given in Table no. 7.

### Table 7: Identification test

| S.N. | Experiment | Observation | Result |
|------|------------|-------------|--------|
| 1.   | Dissolve 50g of the substance under examination in ml of dilute acetic acid or use 1ml of the prescribed solution. Add 10ml water and 0.2ml of 1 M potassium iodine, a yellow precipitate is formed. Heat to boiling for 1 or 2 minutes and allow cooling, the precipitate is reformed as glistening yellow plates. | Lead is present. | Lead is not present. |
| 2.   | To 0.5 ml of a neutral or slightly acid solution of the substance under examination add 0.2 ml of a 0.1 per cent w/v solution of titan yellow and 0.5 ml of 0.1M sodium hydroxide, A bright red turbidity developed which gradually settles to give a bright red precipitate. | Magnesium is present. | Magnesium is present. |
| 3.   | Dissolve 50 mg of substance under examination in 5 ml of water or 5ml of prescribed solution then add 1ml of dilute HCl and 1ml of barium chloride solution white precipitate formed. | Sulphate is present. | Sulphate is present. |
| 4.   | Dissolve 0.1g of the substance in 5ml of water add 0.2 ml of NaOH solution a white ppt are produce add a further 2ml of NaOH solution and ppt solved the add 10ml of ammonium chloride solution the solution remains clear and add 0.1 ml of sodium sulphate solution. A flocculent, white ppt is produce. | Zinc is present. | Zinc is present. |
| 5.   | Dissolve a quantity of the substance under examination containing about 10mg of iron in 2ml of water or use 2ml of the prescribed solution. Add 2ml of H₂SO₄ and 1 ml of 0.1% w/v solution of 1-10 phenanthroline; an intense red colour which is discharge by addition of a slight excess of 0.1M ceric ammonium sulphate is produced. | Ferrous salt is present. | Ferrous salt is present. |
| 7.   | Dissolve 20 mg of the substance under examination in 5 ml of 5 Macetic acid or add 1 ml of glacial acetic acid to 5 ml of the prescribed solution. Add 0.5 ml of potassium ferrocyanide solution, the solution remains clear. Add about 50 mg of on with an ammonium chloride; a white, crystalline precipitate is formed. | Calcium is present | Calcium is present |
| 8.   | Dissolve about 20 mg of the substance under examination in 2 ml of water or use 2 ml of the prescribed solution, add about 0.5 ml of 2 M hydrochloric acid and about 0.5 ml of thioacetamide reagent; no precipitate is produced. Add dropwise 2M sodium hydroxide; a gelatinous white precipitate is produced which redivolves on addition of further 2M sodium hydroxide. Gradually add ammonium chloride solution; the gelatinous white precipitate reappears. | Aluminium is present | Aluminium present |
| 9.   | Flame photometry | Presence of Na and K | Na and K is present. |

• **Rheological study**

The rheological study of Havan ash includes bulk density, tap density, Carr’s index, Hausner’s ratio and angle of repose. Data of rheological study is given in Table no. 8 and 9.

### Table 8: rheological study

| Sr. no | Bulk density | Tap density | Carr’s index | Hausner’s ratio | Flow properties |
|--------|--------------|-------------|--------------|-----------------|----------------|
| 1.     | 0.714        | 0.806       | 11.41        | 1.12            | Good           |
| 2.     | 0.720        | 0.803       | 10.33        | 1.11            | Good           |
| 3.     | 0.714        | 0.806       | 11.41        | 1.12            | Good           |

### Table 9: Angle of repose

| Sr. no | Angle of repose | Properties |
|--------|-----------------|------------|
| 1.     | 40.4°           | Fair       |
| 2.     | 39.7°           | Fair       |
| 3.     | 40°             | Fair       |
Quantitative Analysis

The quantitative analysis of Zinc and Magnesium is done by Atomic Absorption Spectroscopy shown in table no. 10 and the quantitative analysis of Sulphur is done by titration shown in table no. 10.

| Protocol | AAS, Titration |
|----------|----------------|
| Description | Brown Powder |

**Test Parameters**

| Test Parameters | Specification | Observation |
|-----------------|---------------|-------------|
| Zinc            | Not Specified | 182.6 ppm   |
| Magnesium       | Not Specified | 52.6 ppm    |
| Sulphur         | Not Specified | 2.36 %      |

Optimization of Microsponges

- **Particle size analysis of Microsponges**

The data of Particle size analysis of Microsponges is given in Table no. 11

| Sr.no. | Formulation code | Particle size (µm) (mean ± S.D) |
|--------|------------------|---------------------------------|
| 1.     | MSF1             | 42.4 ± 1.23                     |
| 2.     | MSF2             | 40.7 ± 1.54                     |
| 3.     | MSF3             | 38.5 ± 1.26                     |
| 4.     | MSF4             | 31.3 ± 1.25                     |
| 5.     | MSF5             | 37.9 ± 1.19                     |
| 6.     | MSF6             | 42.4 ± 1.23                     |

Figure 7: Microsponges under microscope of formulation MSF5

- **Percentage yield**

The data of Production yield (%) or Percentage yield of Microsponges is given in Table no. 12

| Sr. No. | Formulation code | Production yield (%) |
|---------|------------------|----------------------|
| 1.      | MSF1             | 82.4                 |
| 2.      | MSF2             | 85                   |
| 3.      | MSF3             | 87                   |
| 4.      | MSF4             | 88.7                 |
| 5.      | MSF5             | 91.2                 |
| 6.      | MSF6             | 86                   |
6.3 Evaluation of Gel Base

The results of evaluation of gel base were shown in Table no. 13. The colour of all the formulations was found to be turbid white. All the formulations were stable and has a desired consistency for all the gels. The viscosity and spreadability of the formulations increased with increase in concentration of carbopol 934. The given parameters must be controlled in skin gels included pH (4-9). The pH of all the formulations was within the acceptable range. Human skin is covered with an acid mantle having an acidic pH but due to frequent washing and use of soap the acidity is lost and hence to normalize the skin, gels used should have an acidic range. Considering all the physicochemical parameters, G2 was selected for further loading of microsponges.

| Sr.No. | Parameters   | G1                   | G2                   | G3                   |
|--------|--------------|----------------------|----------------------|----------------------|
| 1.     | Appearance   | Turbid white         | Turbid white         | Turbid white         |
| 2.     | Consistency  | Good                 | Good                 | Good                 |
| 3.     | Grittiness   | -ve                  | -ve                  | -ve                  |
| 4.     | pH           | 5.2 ± 0.047          | 5.5 ± 0.019          | 5.8 ± 0.023          |
| 5.     | Spreadabilty | 9.6 ± 1.005          | 9.8 ± 1.002          | 10.5 ± 1.004         |
| 6.     | Viscosity (cps) | 16120 ± 50.13       | 18191 ± 69.21        | 21532 ± 55.37        |

Table 13: Evaluation of Gel Base

Evaluation of Havan Ash Microsponges Loaded Gel

- Psychometric Evaluation on the Basis of Hedonic Scale

The data of Psychometric Evaluation On the Basis of Hedonic Scale is given in table no. 14

| Sr.No. | Psychometric parameters | F1 | F2 | F3 | F4 |
|--------|--------------------------|----|----|----|----|
| 1.     | Colour                   | 6  | 6  | 8  | 5  |
| 2.     | Odour                    | 7  | 6  | 8  | 6  |
| 3.     | Texture                  | 7  | 6  | 8  | 6  |
| 4.     | Wetness                  | 6  | 7  | 8  | 7  |
| 5.     | Gloss                    | 5  | 6  | 9  | 7  |
| 6.     | Stickiness               | 7  | 6  | 9  | 8  |
| 7.     | Slipperiness             | 6  | 7  | 8  | 6  |
| 8.     | Firmness                 | 8  | 7  | 9  | 7  |
| 9.     | Appearance               | 8  | 7  | 9  | 6  |
| 10.    | Rub-out                  | 7  | 6  | 8  | 6  |
| 11.    | After feel effect        | 6  | 6  | 9  | 7  |
| 12.    | Pick-up                  | 7  | 7  | 9  | 8  |

Table 14: Psychometric Evaluation On the Basis of Hedonic Scale

Evaluation of Havan Ash Microsponges Loaded Gel

- Evaluation of Havan Ash Microsponges Loaded Gel

The evaluation of Havan ash Microsponges loaded gel is done by different parameter. The parameters are Appearance, Consistency, Grittiness, homogeneity, wash ability, pH, Spreadabilty, Viscosity. The data of these all parameter in given in Table no. 15

| Sr.No. | Parameters   | F1            | F2            | F3            | F4            |
|--------|--------------|---------------|---------------|---------------|---------------|
| 1.     | Appearance   | Silver        | Silver        | Silver        | silver        |
| 2.     | Consistency  | Good          | Good          | Very good     | Residue remaining |
| 3.     | Grittiness   | -ve           | -ve           | -ve           | -ve           |
| 4.     | Homogeneity  | Poor          | Good          | Good          | poor          |
| 5.     | wash ability | Poor          | Good          | Very good     | good          |
| 6.     | pH           | 7.2           | 6.5           | 6.3           | 6.9           |
| 7.     | Spreadabilty | 14.3 ± 1.03   | 13.9 ± 0.98   | 14.4 ± 0.777  | 13.5 ± 0.85   |
| 8.     | Viscosity (cps) | 18253 ± 52.41 | 18251 ± 50.12 | 18251 ± 50.12 | 17642 ± 69.37 |
- Stability studies

The prepared gel formulations (F3) were found to be stable upon storage for 6 months, no change was observed in their physical appearance, pH, viscosity, consistency. The result of stability study was shown in Table 16.

| Parameters         | For 0 month          | For 3 month          | For 6 month          |
|--------------------|----------------------|----------------------|----------------------|
| Physical Appearance| Silver               | Silver               | Silver               |
| Consistency        | Good                 | Good                 | Good                 |
| pH                 | 6.3                  | 6.3                  | 6.2                  |
| Viscosity (cps)    | 18251 ± 50.12        | 18251 ± 50.12        | 18251 ± 50.12        |

CONCLUSION

Acne vulgaris remains a common condition in industrialized societies, with many mainstream treatment options available. All these treatments carry risks, and none is completely satisfactory. Natural alternatives are gaining greater research support and have much to offer clinically. A wide range of synthetic therapeutic agents have also been reported to cause acne as their adverse effect. In present hypothesis we have planned to incorporate the herbal Inorganic and Organic actives in the form of Havan ash. The present hypothesis is designed to explore such Natural and traditionally tested active ingredient which act as antibacterial and help to treat acne for better safer natural treatment. As the different elements (like Zn, Fe, Ca, Mg, Na, k, Mn) are present in Havan ash which are helpful to treat acne such as Topical zinc alone as well as in combination with other agents is effective perhaps because of its anti-inflammatory activity and ability to reduce *P. acnes* counts by inhibition of *P. acnes* lipases and free fatty acid levels. Another proposed mechanism for the benefit of zinc in acne is suppression of sebum production by its anti-androgenic activity. Magnesium lowers cortisol production, thereby potential helping to reduce acne by stabilising hormonal imbalances in the body. As a topical acne treatment, sulphur works similarly to benzoyl peroxide and salicylic acid. Sulphur kills the *P. acnes* bacteria, unblocks blocked pores and keeps them clear, and reduces inflammation.

In the preformulation studies of Havan ash include physical description and organoleptic properties, pH, Acid Insoluble Ash, Water-soluble ash, IR Spectroscopy, Identification test, Rheological study, Atomic Absorption Spectroscopy is also carried out.

Quasi-emulsion solvent diffusion is now days the preferred method to prepare porous microparticles. Quasi emulsion solvent diffusion method is simple, less time consuming and involves use of safer ingredients than free radical polymerization and hence more preferred. Eudragit RS100 microsponges containing Havan ash were successfully prepared by this method. The microsponges were prepared (MSF1-MSF5) by quasi emulsion method and were evaluated for its different parameters which revealed many interesting results for efficient preparation of the microsponges. The formulation MSF5 has better results among 6 formulations. MSF5 have its particle size (µm) 31.3 ± 1.25 and production yield (%) 91.2 compared to MSF1, MSF2, MSF3, MSF4, MSF6 they have particle size (µm) 42.4 ± 1.23, 40.7 ± 1.54, 38.5 ± 1.26, 38.5 ± 1.26, 37.9 ± 1.19 and production yield (%) 82.4, 85, 87, 88.7, 86.

Then the Formulation and evaluation of Havan ash microsponges loaded gel is done. The formulation F3 has better results than other F1, F2, F4 formulations. F3 have its appearance silver colour, consistency very good, Grittiness – ve, homogeneity good, wash ability very good, pH 6.3, Spreadability (g/cm/sec) 14.4 ± 0.77 and viscosity (cps) 18251 ± 50.12, have good result on psychometric analysis. Stabilities studies were conducted for the prepared Microsponge loaded gel formulation as per guidelines for a period of 6 months which showed that formulation (F1-F3) were observed stable, while F4 were found unstable due to change in its physical appearance, pH, viscosity, consistency. All these parameters are in optimized range for preparing a controlled release dosage form so showing itself as an optimised formulation in this research work. With the revealed results by different evaluation parameters, it is concluded that microsponges drug delivery system has become highly competitive and rapidly evolving technology and more research are carrying out to optimize cost-effectiveness and efficacy of the therapy.

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