Chemical processing in the stratospheres of the gas giants is driven by incident vacuum ultraviolet (VUV) light. Ethane is an important constituent in the atmospheres of the gas giants in our solar system. The present work describes translational spectroscopy studies of the VUV photochemistry of ethane using tuneable radiation in the wavelength range 112 ≤ λ ≤ 126 nm from a free electron laser and event-triggered, fast-framing, multi-mass imaging detection methods. Contributions from at least five primary photofragmentation pathways yielding CH₂, CH₃ and/or H atom products are demonstrated and interpreted in terms of unimolecular decay following rapid non-adiabatic coupling to the ground state potential energy surface. These data serve to highlight parallels with methane photochemistry and limitations in contemporary models of the photoinduced stratospheric chemistry of the gas giants. The work identifies additional photochemical reactions that require incorporation into next generation extraterrestrial atmospheric chemistry models which should help rationalise hitherto unexplained aspects of the atmospheric ethane/acetylene ratios revealed by the Cassini–Huygens fly-by of Jupiter.

Introduction

Understanding, and perhaps one day exploiting, the environment of extraterrestrial bodies is a central objective of planetary science. The gas giants in our solar system (Jupiter, Saturn, Uranus and Neptune) are rich in molecular chemistry and remain targets of intense scientific study. Like Earth, each of these planets orbits the sun with its own eccentricity and obliquity leading to seasonal variations in incident solar radiation and thus a cycling chemical composition with latitudinal and altitudinal variations in the abundances of the various molecular constituents. Absorption of near-infrared solar radiation by methane (CH₄) makes important contributions to heating the upper atmospheres (stratospheres) of these planets. Methane contributes less to stratospheric cooling, however, which is more reliant on emission from ethane (C₂H₆) and acetylene (C₂H₂). Understanding the balance and interplay between CH₄ and C₂H₆/C₂H₂ is central to understanding the atmospheric dynamics of the gas giants.

Chemical processing in the stratospheres of the gas giants is driven by incident vacuum ultraviolet (VUV) light, even in the distant, gas-poor giants Uranus and Neptune. Numerous possible reactions merit consideration, but common photochemical models for these planetary atmospheres necessarily employ a reduced set pruned from a much larger library of reactions, along with their corresponding rates/branching fractions. These models describe many aspects of the atmospheric chemistry of Saturn and Jupiter reasonably well, but have recognised shortcomings. For example, the dominant C₂H₆ and C₂H₂ generation mechanisms are assumed to involve secondary reactions following photolysis of CH₄. But both the Cassini–Huygens fly-by of Jupiter and terrestrial measurements reveal very different meridional and latitudinal distributions for C₂H₆ and C₂H₂. Such would be surprising if both species are tightly...

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Yao Chang, a Jiayue Yang, a Zhichao Chen, a Zhiguo Zhang, ab Yong Yu, a Qingming Li, a Zhigang He, a Weiqing Zhang, a Guorong Wu, id Rebecca A. Ingle, id Matthew Bain, d Michael N. R. Ashfold, id d Kaijun Yuan, id d Xueming Yang id and Christopher S. Hansen id ft

**Chemical Science**

Ultraviolet photochemistry of ethane: implications for the atmospheric chemistry of the gas giants†

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†Electronic supplementary information (ESI) available: Details of the experimental and computational methods used in this study; ionisation energies and dissociative ionisation thresholds for various CₓHᵧ (x = 1, 2, y ≤ 6) species of present interest; mass spectra showing CH₂ (m/z 14) and CH₃ (m/z 15) products from photolysis of C₂H₆ at λ = 125.6, 121.6, 118.2 and 112.0 nm using two different FEL pulse energies; CH₂+ and CH₃+ ion images analysed to prepare the FT(KER) and δ(TKER) distributions shown in Fig. 4 and a representative C₂H₃+ ion image acquired following λ = 121.6 nm photolysis of a jet-cooled C₂H₆ sample. See DOI: 10.1039/d0sc01746a

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coupled to methane photolysis.\textsuperscript{3,5,10} Neglect of ion-molecule chemistry has been suggested as one possible explanation for this discrepancy,\textsuperscript{4,11,12} but it is also appropriate to question the inputs to the commonly used photochemical schemes. These draw on data\textsuperscript{a} from a range of (often indirect) sources, including predictions, wherein chemical pathways have been included or removed on the basis of how well the model fits the measurements. Ethane is an important participant in these models and, whilst VUV photolysis is accepted as its main destruction route,\textsuperscript{13,14} the dominant fragmentation pathways and photoproducts are not well determined.

Early laboratory studies of C\textsubscript{2}H\textsubscript{6} photolysis at the resonance wavelengths emitted by a xenon lamp ($\lambda = 147.0$ and 129.5 nm) deduced the involvement of (at least) three fragmentation pathways. Two involve loss of H\textsubscript{2} or two H atoms, the other yields CH\textsubscript{4} + CH\textsubscript{3} products.\textsuperscript{15} Subsequent studies using Kr and Ar resonance lamps ($\lambda = 123.6$ and 106.7/104.8 nm, respectively) suggested additional primary fragmentation channels, to CH\textsubscript{4} + CH\textsubscript{3} and, particularly, H + C\textsubscript{2}H\textsubscript{5} products.\textsuperscript{20}–\textsuperscript{28} These studies all involved careful end-product analysis but could not distinguish primary photofragmentation processes from secondary reactions following photolysis, nor yield any dynamical information. More recent imaging studies showed formation of H atoms following C\textsubscript{2}H\textsubscript{6} photolysis at the Lyman-\alpha wavelength ($\lambda = 121.6$ nm, the most intense VUV wavelength in the solar spectrum), with an isotropic velocity distribution peaking at low kinetic energies and a weak tail extending to higher energies. The form of this distribution was attributed to initial C-H bond fission, yielding a fast H atom and an electronically excited C\textsubscript{2}H\textsubscript{5}\textsuperscript{*} fragment, followed by a second (slow) H atom from unimolecular decay of the latter.\textsuperscript{29}

The present study employs two cutting-edge technologies – the intense, pulsed VUV free electron laser (FEL) at the Dalian Coherent Light Source (DCLS)\textsuperscript{28} and an event-triggered, fast framing, Pixel Imaging Mass Spectrometry (PImMS\textsuperscript{2}) sensor\textsuperscript{31} – to advance understanding of C\textsubscript{2}H\textsubscript{6} photochemistry and to identify similarities and differences with the photochemistry of both lighter (i.e. CH\textsubscript{4}) and heavier (e.g. propane (C\textsubscript{3}H\textsubscript{8})) alkanes. The reported data derive from two sets of collision-free experiments: (i) multi-mass velocity-map ion imaging (PImMS\textsuperscript{2} detected)\textsuperscript{32} studies following one-color VUV photolysis of ethane and ‘universal’ (i.e. not quantum state selected) photoionisation of CH\textsubscript{4} and CH\textsubscript{3} photoproducts at four (FEL-produced) wavelengths in the range 112.0 \leq \lambda \leq 125.6 nm, and (ii) VUV photolysis at $\lambda = 121.6$ nm (using photons generated by four wave mixing outputs from a tabletop pulsed laser) and subsequent detection of H atom products using the high resolution H-atom Rydberg tagging technique.\textsuperscript{23,24} The experimental procedures have all been described previously and are thus confined to the ESI.\textsuperscript{†}

Results and discussion

(a) Ethane absorption and the energetics of its various dissociation channels

Fig. 1 shows the chosen photolysis wavelengths superimposed on the electronic absorption spectrum of ethane.\textsuperscript{25,26} As with the other alkanes, the absorption of C\textsubscript{2}H\textsubscript{6} lies in the VUV region but, uniquely amongst the alkanes, its room temperature absorption spectrum displays resolved vibronic structure. This structure is attributed to transitions from the near degenerate highest occupied 3a\textsubscript{1g} and 1e\textsubscript{g} valence orbitals to orbitals with dominant 3p Rydberg character. One or more of these are suggested to have significant antibonding valence $\sigma^*$ character also.\textsuperscript{27} Excitations to the 3s Rydberg orbital in C\textsubscript{2}H\textsubscript{4} are predicted at lower energies, but to be weak – as a result of the molecular center of symmetry – thus distinguishing the 3s Rydberg excitations of ethane from those in CH\textsubscript{4} and the heavier alkanes which all show large absorption cross-sections. This seemingly simple description hides a wealth of potential complexity, however. The degeneracy of the ground (X\textsuperscript{2}E\textsubscript{g}) state of the C\textsubscript{2}H\textsubscript{6}\textsuperscript{+} cation is lifted by Jahn–Teller distortion, and the structure and dynamics of the resulting cation states are further complicated by the energetic proximity of the low lying $A^2A\textsubscript{1g}$ excited state – with the result that even a full interpretation of the threshold photoelectron spectrum of C\textsubscript{2}H\textsubscript{6} remains elusive.\textsuperscript{28} Such interactions must also affect the Rydberg states of current interest – since they share the same ion core(s) – and thus affect the absorption spectrum shown in Fig. 1.

Contemporary computational chemistry methods have enabled global investigations of the ground (S\textsubscript{0}) state potential energy surfaces (PESSs) for species involved in the early stages of the pyrolysis of ethane and other C\textsubscript{1}–C\textsubscript{3} hydrocarbons\textsuperscript{29} but have yet to be directed at the excited state photochemistry of any but the very simplest alkanes. Fig. 2 shows the lower-lying dissociation limits of C\textsubscript{2}H\textsubscript{6}. The predicted minimum energy conical intersections were located using the global reaction route mapping (GRRM) method and are discussed later. The S\textsubscript{0} state correlates adiabatically with the ground state products from either C–C or C–H bond fission (i.e. ground state CH\textsubscript{3} + CH\textsubscript{3} and H + C\textsubscript{2}H\textsubscript{5} fragments). The former is the weaker bond, and the formation of \textsuperscript{13}CH\textsubscript{3} + CH\textsubscript{4} products is attributed to an (essentially barrierless) H atom transfer between the incipient CH\textsubscript{3}...
The energetic thresholds for these three processes are all lower than the calculated barrier to $\text{H}_2$ elimination on the $S_0$ PES ($\approx 5.1 \text{ eV}$). As Fig. 2 also shows, many more spin-allowed fragmentation channels are energetically accessible following electronic excitation of ethane. Table 1 lists no fewer than 17 chemically intuitive channels that require less than the 10.2 eV of energy provided by a Lyman-$\alpha$ photon. Of these, 8, 7, 6 and 5 of the channels yield, respectively, H atoms, $\text{H}_2$ molecules, $\text{CH}_2$ and $\text{CH}_3$ radicals amongst the dissociation products. Such commonalities provide a major challenge for quantitative studies of the primary photochemistry of ethane (and larger alkanes). Of particular relevance to the present study, the reduced models currently used to describe the atmospheric chemistry of Jupiter and Saturn $^1$–$^3$ recognise just reactions (1)–(5) in Table 1.

(b) Ion imaging studies

Fig. 3 shows a representative time-of-flight mass spectrum (TOF-MS) of the ions formed following FEL excitation (at $\lambda = 121.6 \text{ nm}$) of a jet-cooled sample of $\text{C}_2\text{H}_6$ in helium. The spectrum is dominated by a peak associated with $\text{H}^+$ ions. This is unsurprising, given that this wavelength is resonant with the Lyman-$\alpha$ transition from the ground ($n = 1$) state of the H atom. The remainder of the TOF-MS, displayed on a $5 \times$ expanded vertical scale, reveals two clumps of partially-resolved peaks corresponding to $\text{CH}_x^+$ ($x = 2$, 3) and $\text{C}_2\text{H}_y^+$ ($y = 3$–6) ions. The most intense features in the latter are associated with $\text{C}_2\text{H}_3^+$ and $\text{C}_2\text{H}_5^+$ ions. Tables S1 and S2 in the ESI† list relevant adiabatic ionisation and dissociative ionisation thresholds, respectively, and show that four of the neutral products of particular interest (i.e. $\text{CH}_2$, $\text{CH}_3$, $\text{C}_2\text{H}_2$ and $\text{C}_2\text{H}_3$) are amenable to single photon ionisation at 10.2 eV, with dissociative ionisation only a (potential) concern if any of these species carry high levels of internal excitation. Note, however, that the observation of some parent $\text{C}_2\text{H}_6^+$ ion signal highlights the difficulty of completely excluding multiphoton processes even when operating at threshold FEL pulse intensities ($<100 \text{ nJ}$).

The inset to Fig. 3 shows that the relative intensity of the $\text{CH}_2^+$ signal increases as the excitation wavelength is decreased. Note that the data shown in the inset were recorded with the detector sensitivity raised for just the relevant narrow range of mass/charge ($m/z$) ratios, thus allowing averaging over many more acquisitions and improved signal-to-noise ratios. The $\lambda$-dependent trend in the $\text{CH}_2^+$ signal is also recognisable in spectra recorded using higher FEL pulse energies but, as shown in Fig. S1,† the relative peak intensities are also pulse energy dependent. Such variations are not unexpected, given the
Table 1 Possible spin-allowed fragmentation channels for C₂H₆ following absorption of a photon with energy $E_{\text{phot}} < 10$ eV. Reactions (1)–(5) were used to describe the photoinduced loss of C₂H₆ in the recent modelling of the stratospheres of Saturn and Jupiter,¹³ reactions (6) and (7) are implicated in the present data interpretation and the remaining reactions are numbered in order of increasing reaction enthalpy (calculated from data in ref. 47).

| Products | $\Delta H$ (0 K)/eV | Dissociation channel |
|----------|-----------------|---------------------|
| C₂H₄ + H₂ | 1.34 | 1 |
| C₂H₄ + H + H | 5.82 | 2 |
| C₂H₈ + H₂ + H₂ | 3.08 | 3 |
| CH₃ + CH₃ | 4.64 | 4 |
| CH₃ + CH₃ | 3.81 | 5 |
| CH₄ + 1CH₂(|X) + H | 8.55 | 6 |
| C₂H₃ + H | 4.30 | 7 |
| CH₃ + 1CH₂(b) | 5.89 | 8 |
| C₂H₄ + H₂ + H | 6.07 | 9 |
| CH₃ + CH + H | 8.40 | 10 |
| CH₃ + CH + H₂ | 8.40 | 12 |
| 1CH₂(|X) + 1CH₂(|X) + H₂ | 8.81 | 13 |
| CH₃ + 1CH₂(|a) + H | 8.94 | 14 |
| C₂H₅ + H | 9.33 | 15 |
| CH₃ + CH₂ | 9.54 | 16 |
| 1CH₂(|a) + 1CH₂(|a) + H₂ | 9.59 | 17 |

Fig. 3 Time-of-flight mass spectrum of ions formed following photoexcitation of a jet-cooled sample of C₂H₆ seeded in He at $\lambda = 121.6$ nm ($h\nu = 10.19$ eV). Inset: expanded spectra illustrating the $\lambda$-dependence of the m/z 14 (C₂H₃⁺) and m/z 15 (CH₃⁺) peak intensities.

inevitable differences in the wavelength and internal energy dependent photoionisation cross-sections for CH₄, ¹CH₂ and ³CH₂ radicals.

Use of the PImMS² sensor affords not just TOF mass spectra such as those presented in Fig. 3, but also an ion image for each mass channel, in a single acquisition. This provides velocity distributions for each ion peak in Fig. 3. Since C–C bond rupture processes are likely to be pivotal in the cycling of ethane and methane and thus to have a major effect on the atmospheric dynamics, we first present kinetic energy distributions of CH₃ and CH₂ fragments (monitored via the corresponding ions) from the photofragmentation of ethane.

Fig. 4 presents the total kinetic energy release $P(T_{\text{KER}})$ distributions (calculated on the basis that the partner to the observed fragment carries all of the remaining mass) and TKER-dependent best-fit recoil anisotropy ($\beta$) parameters obtained from analysing the ion images retrieved from the selected time slice of the TOF-MS peaks corresponding to (A, B) CH₃⁺ and (C, D) CH₂⁺ ions recorded at FEL wavelengths $\lambda = 125.6$ nm (9.87 eV), 121.6 nm (10.19 eV), 118.2 nm (10.49 eV) and 112.0 nm (11.07 eV). Note that the signal intensities at TKER > 35 000 cm⁻¹ are too low for recoil anisotropy parameters to be fitted satisfactorily. Fig. 4A and C also show the corresponding $P(T_{\text{KER}})$¹² plots (dotted lines) to allow better visualisation of the high TKER data. The raw ion images are shown in Fig. S2 of the ESL.

The distributions derived from the CH₃⁺ ion images (Fig. 4A) assume that the co-fragments are CH₄ (i.e. that the CH₂ fragments derive from reaction (4) in Table 1). This assumption must be correct for the more translationally excited CH₂ products, which display an anisotropic velocity distribution characterised by a positive recoil anisotropy parameter, $\beta \sim +0.5$–0.7 (Fig. 4B), i.e. the CH₂ fragments recoil preferentially along the axis parallel to the polarisation vector $\epsilon$ of the photolysis laser photons. But the $P(T_{\text{KER}})$ distributions extend to TKER $\sim 0$ – implying substantial internal excitation of some of the CH₂ and/or CH₃ fragments. Indeed, as Table 1 shows, the chosen photon energies exceed the thresholds for several three-body fragmentation processes that yield CH₃ products. Some or all of channels (6), (13), (14) and (17) in Table 1 could contribute to the increased low-TKER product yield observed at the two shortest excitation wavelengths — a point to which we return later. Thus the precise form of the $P(T_{\text{KER}})$ distribution at low TKER is ill-defined, since the momentum conservation arguments used to convert a measured CH₂ fragment velocity (derived from the image radius) into a TKER value are likely not to apply in a three-body dissociation. But this does not negate the conclusions that (i) the relative yield of slow fragments in the CH₃⁺ images increases with increasing photon energy and (ii) the slower fragments, which display minimal recoil anisotropy ($\beta \sim 0$), likely arise via one or more of the three-body fragmentation processes.

The distributions derived from the CH₂⁺ ion images (Fig. 4C) peak at TKER $\sim 0$ and show a tail extending to higher TKER that becomes more anisotropic (to positive $\beta$) and relatively more intense as the photolysis wavelength is reduced. As can be deduced from Table 1, the maximum possible TKER of CH₃ fragments formed via reaction (5) following excitation at $\lambda = 121.6$ nm (Fig. 4C) would be $\sim 6.38$ eV (51 500 cm⁻¹); the high-TKER tails of the $P(T_{\text{KER}})$ distributions shown in Fig. 4C (derived assuming C–C bond fission) extend to values for which the direct C–C bond fission channel (5) is the only possible one photon induced CH₄ fragment formation pathway. Most of the imaged CH₃ fragments appear with much lower TKER, however. Table 1 shows several potential sources of slow CH₃ radicals, including three-body dissociations (6), (12) and (14) and the production of an electronically excited CH₃ partner (channel (16)), the relative likelihoods of which are discussed below.
H atom TOF spectra were recorded following photolysis of a jet-cooled C$_2$H$_6$ sample, with the TKER shown in both cm$^{-1}$ and eV (bottom and top axes, respectively). The distributions for each photolysis wavelength are offset vertically for display purposes and the dotted lines in plots (A) and (C) show the square root of $P$(TKER) – normalised to the same maximum value – in order to accentuate the high TKER tails. The TKER$_{\text{max}}$ values associated with the two-body fragmentation channels (4), (5) and (8) as well as the most exoergic three body fragmentations yielding CH$_2$ and/or CH$_3$ fragments (channels (12) and (6)) are also indicated by vertical arrows in panels (A) and (C). TKER$_{\text{max}}$ values for other many-body fragmentation channels can be derived from Table 1 but, as noted in the text, given the assumed TOF to TKER conversion scheme it is likely that the products from such many-body fragmentations would appear at TKER values well below TKER$_{\text{max}}$.

(c) H atom photofragment time-of-flight (TOF) spectra

H atom TOF spectra were recorded following photolysis of a jet-cooled ethane sample in He at $\lambda = 121.6$ nm with $\vec{r}$ aligned at, respectively, $\phi = 0$, 54.7 and 90° to the detection axis and converted to the corresponding $P$(TKER) and $\beta$(TKER) distributions, shown in Fig. 5A and B, by assuming C$_2$H$_5$ as the co-fragment. The fastest products have TKER $\sim 35$ 000 cm$^{-1}$ (~4.3 eV). This TKER value is greater than that reported in the earlier imaging study at this wavelength$^{19}$ but still well below the maximum allowed by energy conservation assuming single C–H bond fission in ethane (channel (7), for which TKER$_{\text{max}}$ $\sim 6.5$ eV). In contrast to the case of CH$_4$, however, the H atom recoil velocity distribution is essentially isotropic.

(d) Active photofragmentation channels

The present work identifies fragments formed by VUV photolysis of C$_2$H$_6$, assures that these arise via collision-free unimolecular dissociation and affords insights into the fragmentation dynamics. The translational spectroscopy data for the CH$_2$, CH$_3$ and H atom products hint at similarities in the fragmentation mechanisms following VUV photoexcitation of C$_2$H$_6$ and CH$_4$ and it is useful to summarise current knowledge of the photofragmentation dynamics of CH$_4$ to provide context for the discussion that follows.

Only the ground (S$_0$) state and a repulsive triplet excited state of CH$_4$ correlate with the lowest energy C–H bond fission limit (associated with H + CH$_3$ products). The first excited singlet (S$_1$) state of CH$_4$ correlates adiabatically with H + CH$_3$ products; the electronically excited CH$_4^*$ fragments predissociate rapidly to H + CH$_2$(a) products.$^{18}$ (Here and henceforth, we use superscript * and $^\#$ symbols to indicate, respectively, electronically and rovibrationally excited products). Nonetheless, experiments find a substantial quantum yield of ground state C–H bond fission products following VUV photoexcitation of CH$_4$, and the...
H atom products display anisotropic recoil velocity distributions – implying that the photoexcited molecules dissociate on a time scale that is much shorter than the rotational period of the parent molecule (which is estimated to be a few picoseconds). These findings highlight the importance of non-adiabatic couplings via conical intersections (CIs) between the S_1 and S_0 PESs. Theory shows that, to form H + CH₂ products, one C-H bond in the photoexcited CH₄ must start stretching and sweep through the plane defined by the other atoms to access the S₀ PES and dissociate. Angular momentum conservation requires that the resulting CH₃ products are highly rotationally excited; indeed, some of these CH₃(X) fragments are formed with so much internal energy that they dissociate further – to H + CH₂ and/or H2 + CH products. Rival distortions have also been identified, whereby photoexcited CH₄ molecules dissociate by eliminating H₂. Theory suggests that the partner at a CI with the S₀ PES (and/or non-adiabatic coupling following photoexcitation of C_2H_6) remain challenging but global reaction route mapping (GRRM) calculations (summarised in the ESIT) can offer important insights by predicting low-lying conical intersections (CIs) between the PESs for the S₀ and S₁ states. The present VUV photoexcitations will populate one or more S₃(n > 1) states of ethane, but we henceforth assume that molecules excited to these higher S₃ states undergo efficient non-radiative coupling to the S₁ state. As Fig. 2 showed, the S₃ state of ethane correlates with H + C₂H₅ and CH₃ + CH₂ products (channels (15) and (16) in Table 1). The C₂H₅ and CH₃ species are both unstable and dissociate to give, respectively, H + C₂H₄ (ref. 39–41) and H + CH₂ (ref. 31) products. As in CH₄, C₂H₅(S₁) molecules can also decompose after non-adiabatic coupling to the S₀ PES. The nuclear distortions required to access the predicted CIs between the S₁ and S₀ PESs (shown in Fig. 2) correlate well with ¹CH₂ elimination once an H atom has inserted between the two C atoms, with C-C or C-H bond fissions and with loss of H₂. The present study is sensitive to the first three fragmentation processes, which are considered in turn. Given the photon energies involved and the multi-dimensional nature of many of these distortions, we can anticipate that (as in the case of CH₄ (ref. 30, 33 and 34)) many of the polyatomic products will be formed with sufficient internal energy that they will fragment further.

**CH₂ radical formation.** The imaging studies reveal CH₂ fragments, with non-zero β values, implying that these are again formed on a timescale shorter than the parent rotational period. The P(TKER) distributions extend to values where the partner fragment can only be CH₄, but not to sufficiently high TKER values to allow unambiguous determination of the electronic state of the CH₂. Spin-conservation arguments suggest that CH₂ radicals formed in tandem with CH₄ will be in their ¹a₁ state (for dissociations that occur following non-adiabatic coupling at a CI with the S₀ PES) and/or ¹b₁ state (for dissociation on the S₁ PES). But the distributions also extend to TKER ~ 0, showing that one or other or both fragments are formed with a broad spread of internal energies. The photoexcitation energies are sufficient to induce three-body fragmentations and, simply on energetic grounds, any of channels (6), (13), (14) and (17) in Table 1 could contribute to signal in the CH₂⁺ ion images. Of these, unimolecular decay of any sufficiently internally excited CH₂⁺ partner would be expected to contribute to the yield of (slow) H and CH₂ products, i.e. the net reaction (14) in Table 1.

Fig. 4A shows an additional feature at low TKER in the distributions derived from the CH₂⁺ images measured at the two shorter wavelengths. This might signify the opening of a new (three-body) route to ³CH₃ products, but this feature more likely indicates the presence of ³CH₂ photoproduction: The ground states of the ³CH₂ radical and the CH₄⁺ cation have very similar geometries. Photoionisation of ³CH₂ thus tends to be vibrationally adiabatic (i.e. to favour Δν = 0 transitions) and, from Table S1,† should only be expected at E_{photon} > 10.39 eV (i.e. at λ < 119.3 nm). Note that the feature at low TKER in the P(TKER) distributions shown in Fig. 4A appears to ‘turn on’ and become more prominent as the photon energy is tuned above this threshold. Several possible routes to forming ³CH₂ products can be envisaged, including the unimolecular decay of highly internally excited CH₂⁺ fragments (from initial C-C bond fission) or of C₂H₅⁺ fragments (following primary C-H bond fission) after non-adiabatic coupling to the S₀ state – as discussed below. Both would contribute to net process (6) in Table 1, though not necessarily exhibit similar energy disposals.

**C-C bond fission.** The tails of the P(TKER) distributions derived from the CH₃⁺ images extend to TKER values that can only be accommodated by assuming C-C bond fission and formation of two CH₂ radicals (i.e. reaction (5) in Table 1). Most of the measured CH₃⁺ fragment velocities imply TKER values far below the maximum allowed by energy conservation, however. Focussing first on the high TKER region in Fig. 4C, the CH₂ fragment yield is clearly rising with decreasing TKER, indicating a preference that one (or both) CH₃ fragments from reaction (5) are formed internally excited. Such energy disposal would likely be a consequence of the nuclear motions that enable non-adiabatic coupling to the S₀ PES. Again, the non-zero β parameter revealed by the CH₃ images imply that these nuclear motions and the ensuing C-C bond fission on the S₀ PES also occur on a timescale shorter than the parent rotational period.

In principle, the entire P(TKER) distribution derived from the CH₃⁺ image could be attributed to channel (5) if the fragmentation dynamics were heavily biased towards forming very highly internally excited CH₂⁺ products. The unimolecular decay of these CH₂⁺ fragments would be a source of the ³CH₂ products inferred above (reaction (6)), and these ³CH₂ products would be expected to display a similar translational energy distribution to that of the CH₂⁺ products (since the light H atom partner would take the bulk of any excess energy released in the secondary fragmentation). Such expectations are consistent with the experimental data and, as noted above, the non-observation of a peak attributable to ³CH₂ products at longer wavelengths (e.g. at λ = 121.6 nm) need not imply that CH₂⁺ fragments are not formed but simply that the ³CH₂⁺ products from their decay are not amenable to photoionisation at the longer wavelengths.
The dominance of translationally 'cold' (i.e. internally 'hot') CH$_3$ products in the $P$(TKER) distributions is striking, however. Table 1 suggests several other potential sources of slow CH$_3$ products. Adiabatic dissociation on the S$_1$ PES to CH$_3$ + CH$_3^*$ products is an interesting contender. This process is exothermic at all wavelengths studied, though the adiabatic S$_1$ PES will likely exhibit a barrier at short $R_{C-C}$ bond extensions as the Rydberg function acquires increasing $\alpha$ antibonding valence character.\(^6\) The CH$_3^*$ radicals would be unstable with respect to H + 1CH$_2$(â) products.\(^{44}$ Again, the H atoms would carry most of any kinetic energy release, so the translational energy distributions of any 1CH$_2$(â) fragments formed in this way should broadly mirror that of their CH$_3^*$ precursor. 1CH$_2$(â) fragments are amenable to photoionisation at all wavelengths investigated in the present work, but the TKER distributions derived from the CH$_2^*$ images measured at the longer excitation wavelengths show no 'spike' at low TKER – suggesting that any contribution to the 1CH$_2$(â) yield from adiabatic dissociation to CH$_3$ + CH$_3^*$ products on the S$_1$ PES must be small compared to that from reaction (4).

C–H bond fission

The $P$(TKER) distribution derived from the H atom TOF measurements (Fig. 5A) extends to TKER values that can only be attributed to prompt C–H bond fission following VUV photolysis of C$_2$H$_6$, i.e. to reaction (7) in Table 1. The C$_2$H$_5$ c-fragments are formed with a very broad spread of internal energies. Analogy with CH$_4$ suggests that this energy disposal is a consequence of the nuclear motions that promote C–H bond fission by non-adiabatic coupling to the S$_0$ PES.\(^{55}$ Most of the 'C$_2$H$_5^*$' products assumed in deriving the $P$(TKER) distribution have sufficient internal energy to dissociate further – by loss of another H atom (net reaction (2)), or H$_2$ (net process (9)), or both (net channel (10)), or to two CH$_2$ species (e.g. via net channels (6) or (11) in Table 1).\(^{44}$ However, the smoothly varying $P$(TKER) distributions shown in Fig. 5A suggest that such overall three- (or more-) body dissociations occur sequentially, i.e. via a prompt C–H bond fission and subsequent unimolecular decay of the resulting C$_2$H$_5^*$ radicals. The low-TKER peak in Fig. 5A could also indicate an adiabatic contribution to the overall dissociation, yielding H + C$_2$H$_4^*$ as primary products – as suggested in the earlier imaging study at $\lambda = 121.6$ nm.\(^{39}$ Any C$_2$H$_5^*$ fragments would dissociate, yielding H + C$_2$H$_4$ products with a spread of translational energies\(^{39,45}$ (i.e. net reaction (2)). Many of the C$_2$H$_4$ products formed by decay of C$_2$H$_5^*$ or C$_2$H$_2^*$ species may well be formed with sufficient internal energy to decay yet further, to H + C$_2$H$_3$ (vinyl) radical products, or by eliminating H$_2$ to yield C$_2$H$_2$. The former products are observed in the present study, via the C$_2$H$_3^*$ peak in the TOF-MS in Fig. 3 and corresponding small ion image shown in Fig. S3 of the ESI,\(^\ddagger\) but the current work is blind to C$_2$H$_2$ products – which were identified by end-product analysis in the early VUV photolysis studies of C$_2$H$_6$ under collisional conditions.\(^{16-18}$ For completeness, we note that C$_2$H$_2$ products could also arise via sequential H$_2$ eliminations from, first, C$_2$H$_4^*$ (formed by non-adiabatic coupling to the S$_0$ state) and then from the resulting C$_2$H$_4^*$ intermediates (i.e. net reaction (3) in Table 1). C$_2$H$_2$ formation by loss of four H atoms from C$_2$H$_6$ is energetically forbidden at the VUV wavelengths of current interest.

The present study does not return quantum yields and, as noted above, is silent regarding some molecular elimination channels. But it certainly identifies several active fragmentation channels and provides new insights into the likely fragmentation dynamics. The present analysis finds no compelling evidence for adiabatic dissociation on an excited state PES – implying efficient non-adiabatic coupling between excited states of C$_2$H$_6$ and to the S$_0$ PES. Many of the present interpretations align with the results of recent quasi-classical trajectory surface hopping calculations for the next larger alkane, propane (C$_3$H$_8$), following excitation at $\lambda = 157$ nm, wherein it was concluded that most dissociations occur after internal conversion to the S$_0$ PES, that the energy disposal in the resulting fragments is governed by dynamical rather than statistical factors, and that the three-body fragmentation processes occur sequentially.\(^{46}$

(c) Implications for modelling the atmospheres of the gas giants

This work provides detailed new insights into the VUV photochemistry of ethane. The results discussed in detail in the above subsections are summarised below in terms of their implications for modelling the atmospheres of the gas giants. These new results must influence future models:

(i) The HRA-PTS studies reveal kinetic energy distributions extending to values that, on energetic grounds, can only be attributed to prompt C–H bond fission, confirming primary C–H bond fission yielding H + C$_2$H$_5$ products (reaction (7) in Table 1). This reaction does not feature in current models used to describe the chemical processing in the stratospheres of the gas giants. Most of the C$_2$H$_5$ species are formed with enough internal energy to decay further, almost certainly yielding some H + C$_2$H$_4$ products. The present study thus supports inclusion of reaction (2) in the modelling and implies that the two H atoms in that case are lost sequentially.

(ii) The kinetic energy distributions derived from the CH$_2^+$ ion images extend to TKER values that can only be attributed to C–C bond fission yielding two CH$_3$ radicals, confirming that the C–C bond fission channel (reaction (5)) is active and supporting its inclusion in the modelling. The finding that the $P$(TKER) distributions peak at TKER ~0 implies that one of the CH$_3$ fragments is generally formed with sufficient internal energy to decay further. If C–C bond fission completes after non-adiabatic coupling to the S$_0$ PES, the resulting CH$_3^*$ fragments most likely decay to H + 1CH$_2$(â) products (i.e. net reaction (6)). This reaction is not included in the current model and, according to the present analysis, will have significantly higher quantum yield than reaction (5).

(iii) The imaging studies confirm formation of CH$_2$ fragments, with a smooth $P$(TKER) distribution that extends to TKER values such that the partner fragment can only be CH$_4$. Spin-conservation arguments and the deduced efficiency of non-adiabatic coupling to the S$_0$ PES suggest that these faster CH$_2$ fragments are formed in the $\tilde{\alpha}$A$_1$ state. The inclusion of reaction (4) in the photochemical modelling is vindicated.
The primary fragmentations and resulting product energy disposals following VUV photoexcitation of ethane are shown to be governed by dynamical rather than statistical factors; three-body dissociations are commonplace and occur sequentially. Clearly, quantitative branching ratios for the various active channels are still needed, but the present work offers several clear pointers. Reaction (7) and, particularly, the three-body fragmentation (6) are active and require incorporation in future modelling. The yield of (currently neglected) reaction (6) is deduced to be larger than that of reaction (5). The processes revealed in this study all involve relatively ‘prompt’ C–H or C–C bond fission, after non-adiabatic coupling to the $S_\text{n}$. PES. As Fig. 2 shows, the respective bond energies are lower than the energy barriers to $C_2H_4$ formation by H$_2$ elimination on the $S_\text{n}$ PES. Analogy with CH$_4$ suggests that any $H_2$ and $C_2H_4$ products formed via process (1) will be both translationally and vibrationally excited. The likelihood that the $C_2H_4^+$ species would have sufficient internal energy to surmount the barrier to eliminating a further H$_2$ (to yield $C_2H_2$ or H$_2$CC) is unclear. We further note that the substantial (~60%) branching into $C_2H_3$ species following VUV photoexcitation of $C_2H_6$ assumed in the current planetary atmospheric photochemistry models derives from indirect measurements made more than half a century ago, and is predicated on an assumption that the decomposition of the internally excited $C_2H_4^+$ species formed via reaction (1) would mimic that deduced following photoexcitation of strategically deuterated ethene (CH$_2$CD$_3$) molecules. Such an assumption must be questionable, given the differences in available energy and the recognised importance of dynamics (i.e. the topographies of, and non-adiabatic couplings between, the PESs sampled in the two cases) in determining the current branching and energy disposal. It seems likely that the current models overestimate the relative yield of $C_2H_4$ (particularly $C_2H_2$) photoproducts.

Conclusions

Translational spectroscopy methods employing two cutting-edge technologies – the Dalian Coherent Light Source (DCLS) Free Electron Laser (FEL) and a fast-framing PlmMS2 camera – have revealed many new insights into the rich (and hitherto largely impenetrable) VUV photochemistry of ethane. The present findings should serve to stimulate ab initio molecular dynamics simulations of this prototypical alkane following photoexcitation at VUV wavelengths and substantial refinements of the models currently used to describe the atmospheric photochemistry of the gas giants. This study (i) concludes that, as in CH$_4$, the VUV photochemistry of ethane is driven by efficient non-adiabatic coupling to, and subsequent direct (and sequential) dissociations on the $S_\text{n}$ PES, (ii) highlights the need to revise current photochemical models of the stratospheric photochemistry of the gas giants – by incorporating the hitherto neglected C–H bond fission channel (7) and the three-body decomposition (6) to CH$_3$ + $^1$CH$_3$ + H products and downgrading the relative yield of primary $C_2H_4$ photoproducts – and (iii) emphasises the pressing need for quantitative product branching fractions. Stratospheric $C_2H_6$ production in the gas giants is driven by VUV photodissociation of CH$_4$, but the present analysis implies that the subsequent photochemical coupling between $C_2H_6$ and $C_2H_2$ is likely to be weaker than currently assumed.

Data and materials availability

The raw ion event lists, H-atom TOF spectra and calculation log files are available from the authors upon reasonable request.

Conflicts of interest

There are no conflicts to declare.

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