Natural Cyclodextrins and Their Derivatives for Polymer Synthesis
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Cyclodextrins (CDs) are a family of cyclic oligosaccharides with a hydrophilic exterior surface and a nonpolar cavity interior, therefore, CDs can form inclusion complexes through noncovalent interactions with a broad range of hydrophobic guests. CD derivatives and CD-based polymers find important uses in different fields such as pharmacy, cosmetics, biomedicine, textiles, and food domain due to their unique properties during the past decade. Hence, in this review, functionalized CDs and CD-based polymers are classified and discussed according to their synthetic approaches comprehensively to help polymer chemists for the development of new CD-based materials for different types of applications.

1 Introduction
Cyclodextrin (CD) is a cyclic oligosaccharide with a bucket-like shape. The hydrophobic cavity of cyclodextrin can encapsulate small hydrophobic molecules and can be utilised in the design of supramolecular structures.1 Cyclodextrin is a cheap, widely available, biocompatible and biodegradable material. Valued for encapsulation properties, cyclodextrin found applications in many fields such as pharmacy,2–5 personal care products,6–7 biomedicine,8–10 food,11–13 molecular recognition14,15 and supramolecular chemistry.16–20 Cyclodextrin also proved to be an extremely versatile molecule for polymer science.16,21,22 It is particularly valued for easy functionalisation, including attachment of initiator moieties as well as unsymmetrical modification.21 Cyclodextrin is compatible with a range of polymerisation techniques such as ATRP (Atom transfer radical polymerisation),23–25 ROP (Ring Opening Polymersisation),26,27 and RAFT (Reversible Addition-Fragmentation Chain-Transfer) polymerisation.28,29 Its water solubility and biodegradability make it a perfect candidate for ‘green polymerisation’.30–31 Hydroxyl groups not directly involved in the polymerisation can be later utilised in post-polymerisation modifications to tailor the product for advanced application. If chains are grown from just one face, the other face can be conjugated with a drug or targeting molecule forming hybrid structures that can be applied in cancer therapy and bioimaging.32 Cyclodextrin can also play a ‘supportive’ role in polymer synthesis, for example, cyclodextrin form inclusion with monomer thus improving its solubility and control over polymerisation reaction.33,34 In emulsion polymerisation, cyclodextrin found application as a phase transfer agent35 and in troublesome aqueous RAFT polymerisation, it can solubilise chain transfer agent.36

There are several comprehensive reviews on cyclodextrin-based polymer materials37–40 including materials based on host/guest interactions41 and polyrotaxanes42. A huge amount of work has been done in the field of cyclodextrin-polymer conjugates and this can be overwhelming for beginners. This review is aimed at the newcomers who would like to get an understanding of the opportunities that cyclodextrin brings but also its shortcomings and common pitfalls. The review has four sections: Introduction, Functionalisation of CD, CD-polymer covalent conjugates and Applications. More than 10,000 derivatives of α-, β- and γ-CD have been reported43 which vastly exceeds possibilities of this review. The scope of CD derivatives has been narrowed down to the toolbox of the most common functionalities in the field of CD such as tosylate, halogens or alkyl groups. Particular attention is paid to the derivatives useful for specific applications in polymer chemistry such as CD-initiators or monomers. For more in-depth review on derivatives of cyclodextrins, the reader is directed elsewhere.43,44 The section on covalent conjugates will give an overview of how different polymerisation techniques can be utilised for the synthesis of polymers with certain architectures. The last section will cover representative applications of CD-polymer covalent conjugates in various fields such as preparation of nanoparticles, gene delivery and hydrogels.

There are three cyclodextrins commonly used in polymer chemistry: α-CD, β-CD and γ-CD which are built of six, seven and eight glucopyranose units accordingly, connected via α,1,4-glycosidic bonds (Figure 1).45 Hydroxyl groups are located on the edges of the cyclodextrin bucket which is in turn made of the sugar backbone and glycosidic oxygen bridges. By convention, primary hydroxyls are at the top and secondary hydroxyls at the bottom of the structure. While secondary hydroxyls form strong hydrogen bonds rigidifying the bottom of the cyclodextrin, primary hydroxyls are free to rotate thus reducing the diameter of the top face and giving cyclodextrin truncated cone-like shape.46 β-CD is the most applied cyclodextrin due to its availability however it has also been reported to exert more...
potent cytotoxicity which is an important factor for potential bio applications. 

β-CD is rather rigid and forms a complete set of six intramolecular H-bonds, making it the least water soluble out of all known cyclodextrins. To increase its water solubility, which is particularly important for encapsulation of hydrophobic molecules in aqueous solvent, some of the hydroxyl groups can be converted to sulphate groups. Bigger cyclodextrins such as γ- or δ-CD are no longer symmetrical and as they collapse upon themselves, the cavities become smaller rather than bigger and the complexing capacity of various guest is diminished.

There are three hydroxyl environments with distinct reactivity, seven hydroxyls each, giving twenty-one in total. Primary hydroxyls at C6 position are most basic and nucleophilic, and hence most reactive. Secondary hydroxyls at C2 position are most acidic and those at C3 position are most inaccessible due to steric hindrance and hydrogen bonding and hence least reactive (Figure 2). 

2 Functionalisation of CD

Preparation of CD-polymer conjugates starts with the functionalisation of cyclodextrin. The choice of functionality and degree of substitution depends on the polymerisation technique and desired polymer architecture. Substitution of all hydroxyl groups can give a multiaromatic initiator for the synthesis of star polymers. Monosubstituted cyclodextrin can be utilised in the preparation of CD-functionalised monomers which upon polymerisation will give polymers with cyclodextrin side chain functionality. Primary and secondary faces can be modified selectively thus enabling for the formation of asymmetric multiaromatic constructs. Restricting the extent of the reaction to just one face of cyclodextrin is possible by leveraging the aforementioned differences in the reactivity of hydroxyl groups. However, one needs to take an account on other factors too, such as complexing capabilities of cyclodextrin. Encapsulation of solvent or chemical reagent can have a profound influence on reaction rate and selectivity. The solvent can even switch selectivity as it has been reported for monotosylation. Complexation implies certain orientation of the guest molecule and can give access to hindered C3 hydroxyls, which are difficult to modify selectively. Such an approach led to the successful attachment of cinnamyl functionality. Further, the chemistry of by-products of reaction needs to be considered as cyclodextrins are unstable under acidic conditions and a base such as pyridine or imidazole is often required to scavenge protons.

According to D’Souza et al. protocols for modification of cyclodextrin can be classified into one of the three categories: (a) clever reactions exploiting aforementioned differences in reactivity; (b) long methods involving several protection-deprotection steps; (c) ‘sledgehammer’ indiscriminate reactions with lengthy purification procedure. Sometimes no clever methods are available and indiscriminate reactions with lengthy purification procedures have to be followed. There is ongoing research on more efficient and greener synthetic methods such as mechanochemical synthesis under solvent-free conditions using a planetary ball mill. Many protocols were established before modern state-of-art analytical facilities were available and selectivity of reported reactions as well as purity of the final product were subsequently questioned or even disproved.

Reactive electrophiles will attack both rims of the cyclodextrin bucket indiscriminately. Nevertheless, persubstitution usually requires an excess of the reagent and often a mixture of products with varying degrees of substitution is obtained. C-2 and C-3 substituted derivatives and complicated substitutions patterns are often not achievable without protecting groups. Any standard alcohol protecting group can be used as long as it does not require a high concentration of strong acid for deprotection (Table 1).

Table 1 Methods of introduction and cleavage of common protecting groups in the synthesis of cyclodextrin derivatives.

| Position | Protecting group | Method of introduction | Cleavage | Reference |
|----------|------------------|------------------------|----------|-----------|
| C6       | Silyl ether      | TBDMS                  | BF₃ or TBAF | 54,55     |
| C6       | Benzyl          | Benzyl chloride        | Hydrogenolysis | 56,57     |
| C2&C3    | Methyl*         | Methyl iodide          | NaOMe/MeOH | 58        |
| C2&C3    | Acetyl*         | Acyl chloride          | NaOMe/MeOH | 28,59     |

*Requires prior silylation and subsequent desilylation of the primary face.
Tert-butyldimethylsilyl (TBDMS) is an excellent protecting group for 6-position and can be cleaved with BF$_3$ in tetrahydrofuran or tetra-n-butylammonium fluoride. Alternatively, primary hydroxyls can be benzylated with benzyl chloride and deprotected via palladium catalysed hydrogenolysis. The remaining hydroxyl groups can be acetylated prior to the polymerisation to increase the solubility of cyclodextrin derivative in common organic solvent. After polymerisation, acetates can be cleaved with NaOMe/MeOH.

In a similar way, cyclodextrin can be methylated to improve its water solubility – per-O-methyl-β-CD is 10-fold more soluble than native CD. Such approach is of particular importance in aqueous polymerisation and methylated-CD has been for example used as part of a supramolecular pH-sensitive photoinitiator. Similarly, water soluble copolymer with methylated CD and proline has been used as a nanoreactor to mimic enzymes and provide hydrophobic environment for the reaction yet without affecting homogeneity of the reaction solution. Random hydroxypropylation also improves water solubility of CD. This comes as a result of disruption of the tight hydrogen bonding network formed by cyclodextrin hydroxyl group.

**Table 2 Solubility of common CD-derivatives.**

| Derivative                                  | Solubility                                      |
|--------------------------------------------|-------------------------------------------------|
| Native β-CD                                | Moderate water solubility$^{45}$                 |
|                                            | Soluble in DMF, DMSO$^1$                        |
|                                            | Low solubility in NMP$^1$                       |
| Methylated β-CD (various substitution patterns) | Good water solubility (maximum solubility for 13-14 methoxy per CD$^{43}$) |
| Hydroxypropylated β-CD (various substitution patterns) | Good water solubility$^{60}$ |
| Sulfated β-CD (various substitution patterns) | Good water solubility$^{60}$ |
| Peralkylated β-CD                          | Soluble in common organic solvents$^{44}$       |
| Per-6-halogeno-per-6-deoxy-β-CD             | Soluble in polar organic solvents (DMF, DMSO)$^{44}$ |
| Perallylated β-CD                          | Soluble in common organic solvents$^{48}$       |
| Per-6-thio-per-6-deoxy-β-CD                 | Poor solubility in acetone, MeOH, chloroform, tetrahydrofuran Good water solubility in polar organic solvents – DMF, DMSO$^{27}$ |
| Heptakis[2,3,6-tri-O-(2-bromo-2-methylpropionyl)]-β-CD | Soluble in common organic solvents such as ether, toluene, dichloromethane except aliphatic hydrocarbons, low water solubility$^{23,65}$ |

Although there is literature available concerning water-soluble CD derivatives, no systematic study was performed to assess solubility in common organic solvents which are usually more suitable for polymerisation than water. In general, adding alkyl, silyl and acetyl groups improve solubility cyclodextrin in organic solvents. Information on solubility of common CD derivatives is presented in Table 2.

### 2.1 Substitution of primary hydroxyls

The common functionalisation starting points are halogen and tosylate cyclodextrin derivatives (Figure 3) Both functional groups can be smoothly displaced via nucleophilic substitution. Whereas mono-tosylation renders an excellent precursor for CD-based monomer, pertoxylation of the primary face is a poor choice as the reaction suffers from the formation of 6-anhydro product. One of the first reported methods for monotosylation used p-toluene sulfonyl chloride to give 6-O-mono-6-(p-toluenesulfonyl)-6-deoxy-β-CD but required chromatographic purification. Reaction with p-toluenesulfinic anhydride, on the other hand, did not require purification by chromatography and gave the product in 61% yield; unreacted anhydride was conveniently removed by filtration.

![Figure 3 Methods of substituting cyclodextrin hydroxyl groups with tosylate or halogen to give common synthetic intermediates.](image)
More recently, preactivation of tosylate with imidazole was found to improve selectivity towards 6-o-mono-p-toluenesulfonyl-6-deoxy-β-CD product although other substitution patterns are still observed.68 The first protocols for halogenation used methanesulfonyl halides or triphenylphosphine with iodine gas.57,69,70 Later, more convenient route using N-halosuccinimides was developed.74 This approach is currently favoured as it uses a stable reagent and works for different halogens. Heptakis(6-deoxy-6-halogeno)-CDs are soluble in polar solvents such as pyridine, DMF (dimethylformamide) or DMSO (dimethyl sulphoxide) but their solubility in nonpolar solvent can be increased by esterification (for example acetylation) of the remaining secondary hydroxyls.44

Both halogen and tosylate can be displaced by a nucleophile (Figure 4). Reaction with excess of sodium or lithium azide provides mono-6-deoxy-6-azido-CD, an excellent partner for click chemistry, in quantitative yield.68,72–74 Alternatively, azido-CD can be prepared via Vilsmeier-Haack type reaction directly from native cyclodextrin with lithium azide, triphenylphosphine and carbon tetrabromide.75 This bypasses the longsome tosylation. Similarly, halogen can be displaced by thiourea to give thiol-CD, useful in thiolene reaction. All reported protocols require large excess of thiourea and prolong reflux.76–79 A different method developed by Marsura et al. is based on Mitsunobo reaction where thi-CD can be obtained directly from native cyclodextrin by reaction with a thiol, disopropyl azodicarboxylate and triphenylphosphine.80

Azide group can be also reduced with triphenylphosphine75 or palladium black under hydrogen gas66 to give amino-CD. For other amines, the direct substitution of tosylate with liquid amine is suitable.81,82 More recently, microwave assisted method was developed by Milton et al. that significantly shortened reaction time compared to standard procedure for tris(2-aminoethyl)amine from 48h to just half an hour.83 Amine derivatives found its use in the preparation of more complicated CD-based compounds via amine-carboxylic acid coupling. With the aid of DCC (N,N’-Dicyclohexylcarbodiimide), TEMPO (2,2,6,6-Tetramethyl-1-piperidinyloxyl) derivative of benzoic acid was coupled with heptakis(6-deoxy-6-amino)-β-CD to yield TEMPO-CD for controlled polymerisation.84

2.2 Substitution at C-2 and C-3

Substitutions at 2- and 3-position are less common due to the cumbersome synthesis hindered by the high reactivity of C6 hydroxyls. Selective substitution without protecting groups is often impossible or requires tedious purification. To selectively functionalise C2 hydroxyls, they should be first deprotonated to form 2-alkoxides that are suitable for nucleophilic attack.75,85 Extra care must be taken when choosing a solvent for such reaction as solvent can strongly influence nucleophilicity of oxanions. Even though C2 hydroxyls are the most acidic, equilibration time is required to avoid deprotonation at other positions. Such approach allowed Jurczak et al. to prepare cyclodextrin with an allyl group.86 After deprotonation with lithium hydride and 24h of stirring, propargyl bromide was added to yield mono-2-propargyl-β-CD at 38% yield which is very satisfactory for substitution at 2-position (Figure 5). Monotosylation at C2 was also reported. Ueno and Breslow used 3-nitrophenyl toluenesulfonate as sulfonating agent but the product was isolated at only 10% yield.87 Formation of alkyltin alkoxide C2 prior to tosylation with p-toluene-sulfonyl chloride improved the yield to 29%.88 Both approaches require purification on column chromatography. Once sulfonate is incorporated it can be displaced by nucleophiles in an analogous way to mono-6-deoxy-6-tosylate-CD e.g. with sodium azide.89 To pertosylate C2 hydroxyls, the primary group has to be protected, for example with t-butyl dimethylsilyl groups which can be cleaved with boron trifluoride.90

C3-position is extremely difficult to modify selectively. Necessary protection of both 2- and 6-positions leads to such a steric crowding that accessibility of 3-positions is too low to react. A promising approach is selective deprotection such as monodesilylation.91

2.3 Persubstitution

Persubstitution i.e. substitution of each hydroxyl group by the same functional group is achievable for a limited set of reagents. Even reactive reagents can struggle as the steric hindrance increases with successive substitutions. In a presence of a base, reactions with alkyl or allyl halogenide proceed smoothly, such as reaction of β-CD with methyl iodide to give per-O-methylated-CD with 74% yield.92

Figure 4 Both tosylate and halogen can be easily displaced by nucleophiles to give synthetically useful cyclodextrin derivatives.

Figure 5 Incorporation of propargyl and allyl group at C-2 position.
Achieving good yields for allylation is more difficult and the first attempt on the synthesis of heptakis(2,3,6-tri-O-allyl)-β-CD used allyl iodide but the reported yield was only 32%.92

Recently, perallylated α-, β- and γ-CDs were obtained by reaction with allyl bromide and sodium hydroxide under argon atmosphere in 4h in quantitative yield and required only minimal purification.93 As the allylation reaction is exothermic, low temperature (0°C) of the reaction is crucial to ensure good yield and avoid incomplete allylation.93 Similarly, cyclodextrin can be easily peracetylated with acetic anhydride although this intermediate has little use in polymer chemistry.94 Peracetylation can be catalysed by base95 or acid.96 Persubstitution by esterification is of particular interest for polymer chemists as it allows for the incorporation of ATRP initiator moieties.97 Whereas some esterification such as perbenzoylation with benzoyl chloride in pyridine are reliable giving almost quantitative yield,98 other result in various degrees of substitution99 as discussed in detail in later.

Another strategy to obtain fully functionalised cyclodextrin is to modify two faces separately, with different reagents. Liu and co-workers reported asymmetrically functionalised cyclodextrin with azide functionality on the top face and ATRP initiator at the bottom.99 Haddleton et al., however, following a similar protocol obtained only partial esterification giving (N(H)β-CD-(Br)10 as confirmed by MALDI-TOF MS (Matrix-Assisted Laser Desorption/Ionization-Time Of Flight Mass Spectrometry).39

Persubstitution can also take place at a specific group (C6, C2, C3) or combination of thereof. As highlighted in section 2.1 primary hydroxyls can be substituted with halogens providing heptakis(6-deoxy-6-halogeno)-CD – useful synthetic intermediates. Primary hydroxyls cannot be selectively alkylated or esterified directly. Instead, a sequence of protection-deprotection steps must follow: the primary face can be silylated, then the secondary face esterified. After detrination of the primary face it can be reacted with a respective alkyl halide under basic conditions.100 Analogous path was designed for esterification - pivaloylation.101 Similarly, peralkylation of C2 position cannot proceed directly and requires protection of the primary face.100 BaO allows for selective allylation of position 2 and 6 giving heptakis(2,6-O-diallyl)-β-CD (Figure 5).102 Simple protection of primary face will lead to perallylated cyclodextrin at position 2 upon reaction with allyl halogenide in the presence of BaO.

### 2.4 CD-functionalised monomers

CD-based monomers can be directly polymerised to obtain polymers with a high density of pendant cyclodextrin groups. Monomer has to be prepared in a well-controlled manner or otherwise extensive purification is required to avoid multivinyl-CD monomer. Such impurity could lead to the formation of branched or crosslinked polymers.

The first-reported CD-monomer was based on acrylate and was prepared from the native cyclodextrin with m-nitrophenyl acrylate.103 To avoid side reactions, mild conditions and short reaction times were applied, thus limiting the yield of the reaction to 20%. Selectivity was increased by the inclusion of m-nitrophenyl ester in cyclodextrin cavity. Similarly, other acrylate-based monomers were synthesised from native cyclodextrin.104 Such transesterification reactions are likely to lead to undesired di- and higher substitution products. To achieve higher synthetic precision, Liu et al. used monosubstituted cyclodextrin. Ethylenediamine-modified cyclodextrin was prepared from mono-6-(p-toluenesulfonyl)-6-deoxy-β-CD and reacted with glycidyl methacrylate (Figure 6).105 This approach is often used nowadays and works well with different amine-CD such as piperazine-modified cyclodextrin.106

To increase coupling efficiency, ‘click’ chemistry can be employed such as CuAAC (Copper-catalyzed azide–alkyne cycloaddition) coupling. Recently, methacrylate modified CD (mono-(1H-1,2,3-triazol-4-yl)(methyl)2-methylacryl-β-cyclodextrin) was synthesised by means of azide–alkyne cycloadditions of propargyl methacrylate and mono-6-azido-6-deoxy-β-cyclodextrin. Reaction proceeded with full conversion.107 If microwave radiation is used, the reaction time is reduced from 24h to 30min.108

#### 2.5 Incorporation of initiator groups for multiarm polymers

Since the development of controlled polymerisation techniques, the precise engineering of polymer architectures became possible, among them multiarm polymers. Multiarm star-shaped polymer consists of linear arms branching from a core. Such polymers attracted special attention of both academia and industry due to their low viscosity, high arm density and degree of surface functionality. Cyclodextrin is suitable for both core-first and grafting-to approach (Figure 7).

As cyclodextrin can be asymmetrically functionalised with orthogonal initiating functionalities, miktoarms can be prepared.

Cyclodextrin allows for unusually high grafting densities as for typical well-defined multiarm initiators. Primarily, 7,48,108 or 2153 sites are accessible but intermediate numbers were also reported.111,112

![Figure 6](https://via.placeholder.com/150)

**Figure 6** The two most common approaches in preparation of CD-based monomers. A: amine-CD is reacted with glycidyl-methacrylate. B: Copper mediated azide-alkyne coupling.
Arms not directly involved in the polymerisation can be further modified to tailor the solubility of the initiator or attach various pendants such as sugars. To leverage synthetic precision of controlled polymerisation methods, the initiator has to be well defined and number of functional sites known. Knowledge of the exact degree of substitution is needed to correctly calculate the ratio of reagents for the subsequent polymerisation reaction. The proper characterisation is therefore crucial and MALDI-TOF MS analysis should be carried out to support the structure of the synthesised initiator.

2.5.1 Esterification

Majority of CD-polymer conjugates reported in the literature were prepared via Atom Transfer Radical Polymerisation (ATRP) with initiator moieties attached via esterification of cyclodextrin with alpha-halocarboxylic halide. The first reported method used 2-bromoisobutyric anhydride in pyridine with a catalytic amount of 4-(di-methylamino)pyridine to give heptakis[2,3,6-tri-O-(2-bromo-2-methylpropionyl)]-β-CD. However, despite a long reaction time, poor yield was achieved. The use of 2-bromo-isobutyryl bromide significantly improved the yield: from 17% to 66.8% or even 89.5% when dialysis purification was used. To avoid the formation of pyridinium bromide precipitate that would slow down the reaction and limit the yield, 1-methyl-2-pyrrolidione (NMP) was used instead of pyridine. This approach is universal and also works with different acyl halides such as 2-bromopropionyl bromide.

However, the full esterification was subsequently disproved by two independent research groups which reported only 16 or 14 average degrees of substitution, as confirmed by MALDI-TOF.

In theory, the extent of esterification can be tuned by varying reagent ratio to limit the extent of esterification, as summarised in Table 3. Nevertheless, the protocols are often not reproducible and insufficient characterisation of the product is given; without proper mass spectrometry analysis, the degree of substitution remains questionable. Hence, using protecting groups seems to be a more reliable approach. The primary face can be protected as silylates and secondary face as methyl ethers or acetylates. Often, particularly for silyl groups, acid scavenger such as imidazole is needed to avoid deprotection of hydroxyls by acidic by-product of esterification. Even though protection-deprotection requires additional steps, it facilitates purification and can speed up the synthesis.

Although bromine is the predominant halogen in the literature, chloro-CD initiators can also be successfully prepared. Analogously to bromo derivative, chloro initiator was prepared via esterification with 2-chloropropionyl chloride. Iodine-CD-based initiator was synthesised for iron-based ATRP. As the direct esterification with iodoisobutryl bromide was unsuccessful, halogen exchange of octadeca-O-(isobutryl bromide)-α-CD was implemented.

2.5.2 Conjugation chemistry

Due to the aforementioned pitfalls, alternatives to esterification were examined. Becer et al. investigated three click reactions: CuAAC, Michael thiol addition and radical thiol-ene reaction. The first two approaches suffered from side reactions. Copper reaction gave higher molecular weight species which were assigned to be a result of radical-radical coupling as conditions of the click reaction resembled ATRP set-up. Michael addition with 2-(2-bromoisobutryloxy) ethyl acrylate also led to side products due to the competitive thiol-bromo reactions. The last reaction, radical thiol-ene with allyl 2-bromo-2-methylpropionate was proved successful although SEC (Size Exclusion Chromatography) showed minor dimer formation. This allowed for the preparation of an initiator with 7 active sites (Figure 8). This photo-click reaction took place at room temperature, was completed in 5h and did not require protection of secondary face or chromatographic purification, thus rendering click reaction a superior technique to esterification. Reaction progress can be easily monitored with NMR (Nuclear magnetic resonance) as clear disappearance of allyl proton peaks is observed. The remaining disadvantage is oxidative instability of thiols and necessity for the reduction of thio-CD with dithiothreitol (DTT) prior to the reaction to remove.
**Table 3** Summary of Br-CD ATRP initiators with various degrees of substitutions prepared without protecting groups.

| Degree of substitution | Yield [%] | Reagent and stoichiometry wrt. OH | Wrt to CD | Solvent; other reagents | Reaction conditions | Characterisation | Ref. |
|------------------------|----------|----------------------------------|----------|-------------------------|---------------------|-----------------|------|
| 1                      | N/A      | ![Initiator 1](image) biBB1:4.8  | 4.4      | DMF                     | 1) 2 hr at 0°C, 2) 12 hr at RT, N$_2$ atmosphere | NMR | 116 |
| 3                      | N/A      | BiBB 1:5.25                      | 4        | DMF                     | 1) 2 hr at 0°C, 2) 12 hr at RT, N$_2$ atmosphere | NMR | 117 |
| 5                      | 78.8     | BiBB 1:3                        | 7        | Pyridine, chloroform    | 24 hr at RT, N$_2$ atmosphere | NMR + FTIR + Elemental analysis | 118 |
| 6                      | 48       | BiBB 1:3.4                      | 16.2     | NMP                     | 0°C->RT, 12h | NMR | 119 |
| 7                      | N/A      | BiBB 1:3                        | 7        | DMF                     | 1) 4h, 0°C, 2) 30h, 40°C | NMR, FTIR | 120 |
| 8                      | 82.64    | BiBB 1:2.6                      | 8        | NMP, TEA, DMAP          | 2) 24 hr at RT, N$_2$ atmosphere | NMR | 112 |
| 10                     | 93       | BiBB 1:2                        | 10       | NMP, TEA, DMAP          | 2) 24 hr at RT, N$_2$ atmosphere | NMR | 121 |
| 14                     | 75       | BiBB 2:1                        | 41.4     | NMP                     | 1) 2 hr at 0°C, 2) 22 hr at RT | NMR + MALDI-TOF MS | 113 |
| 16                     | 68.5     | BiBB 3.7:1                      | 78.3     | NMP                     | 1) 2 hr at 0°C, 2) 56 hr at RT | NMR + MALDI-TOF MS | 99 |
| 18                     | N/A      | BiBB1:1                         | 21       | NMP                     | 2) 22 hr at RT, N$_2$ atmosphere | NMR | 122 |
| 21                     | 17       | ![Initiator 2](image) 3:1       | 64       | Pyridine, DMAP          | 96 hr at RT | NMR + FTIR + Elemental analysis | 21 |
| 21                     | 66.8 or 89.5* | BiBB 2:1 | 42 | NMP | 1) 2 hr at 0°C, 2) 18 hr at RT | NMR + FTIR | 65 |
| 21                     | 21       | ![Initiator 3](image) 3:1       | 64       | NMP                     | 96 hr at RT | NMR + MALDI-TOF MS | 78 |

*Dialysis purification; RT = room temperature, TEA = trimethylamine, DMAP = 4-Dimethylaminopyridine, FTIR = Fourier transform infrared spectroscopy.
any disulfide bridges. The reduction can be bypassed by switching functional groups of the conjugating partners as HS-EBiB is stable under storage conditions and can be readily reacted with allyl-β-CD.54

Increased solubility of allyl-CD as compared to native CD allowed for reaction to be run in common organic solvent rather than NMP which is difficult to remove. This thiol-ene reaction gave heptakis[2,3,6-tri-O-6-(3-thiahexyl)-2-bromo-2-methylpropanoate]-β-CD (21-Br-S-β-CD) in 90% yield and the product was duly analysed with MALDI-TOF. To ease data analysis and remove unnecessary isotopic patterns, bromine atoms were removed with tributylthine hydride. 7 and 14-active site species should also be available although this would require a protecting group, for example, benzylation of the secondary face.

2.5.3 Chain transfer agents

Efforts to employ RAFT polymerisation were also undertaken. In a similar manner to ATRP initiators, CD-coupled RAFT agents were prepared. The first study reported functionalisation of just the primary cyclodextrin face with a RAFT agent to avoid steric hindrance around trithiocarbonate moiety (Figure 9).28 Unmodified cyclodextrin was reacted with 3-benzylsulfanylthiocarboxylsulfanyl-propionic acid chloride to give β-CD-RAFT with six functional groups (Heptakis[2,3-di-O-acetyl-6-O-(3-benzylsulfanylthiocarboxyl-sulfanyl-propionyl]-cyclomaltoheptaose). The remaining hydroxyls were acetylated to improve the solubility of the chain transfer agent in styrene whereas modification with 2-hydroxyethyl acrylate increased solubility in an aqueous medium.123 Cyclodextrin formed a part of the Z group of the RAFT agent so it should not fragment during the subsequent polymerisation reaction. As an alternative to esterification, RAFT agent can be attached via DCC coupling of amino-CD with trithiocarboxylic acid.124 Mono-RAFT-alpha-CD could act as a supramolecular catalyst by complexing monomer inside the hydrophobic cavity.

In summary, cyclodextrin can be modified with tosylate group or halogens to give robust synthons which can be further utilised in the synthesis of initiators or used as a place of attachment for polymer chains. TEMPO, ATRP initiator and RAFT agent have been successfully prepared starting from either pre-functionalised or native cyclodextrin. The primary or secondary hydroxyl group can be utilised selectively either by exploiting specific reactions or incorporation of standard protecting groups.

3 CD-polymer covalent conjugates

There are three basic architectures of covalent cyclodextrin-polymer conjugates: multiarm CD-centred polymers, CD-pendant polymers, and CD-capped polymers (Figure 10). All three can self-assemble to form sophisticated structures. In this section, different polymerisation methods and their compatibility with cyclodextrin will be discussed. Native cyclodextrin has not been reported to interfere with polymerisation as such however cyclodextrin should be thoroughly dried prior to any reaction. As ionic polymerisation is nucleophile-sensitive, naked cyclodextrin face which is not involved as initiators should be protected.

3.1 Multiarm polymers cyclodextrin-centred polymers

3.1.1 Prepared via ATRP

CD-based initiator can be employed for both conventional23,64,125 and variations of ATRP such as single-electron transfer living radical polymerisation (SET-LRP)99,109, electrochemically mediated ATRP (eATRP)24,126 and activator regenerated by electron transfer (ARGET) ATRP127. Conditions for ATRP polymerisation using CD-based initiator are similar to the standard set-up and a range of monomers have been successfully polymerised. As for multiarm polymers, attention has to be paid to optimise conditions to minimise star-star coupling.

Haddleton et al. reported the first polymerisation from a cyclodextrin core with attached ATRP initiators.23 heptakis[2,3,6-tri-O-(2-bromo-2-methylpropionyl]-β-CD (21-Br-β-CD) was used to polymerise methyl methacrylate (MMA), styrene and to prepare block copolymers by sequential addition. The reactions were run in toluene using standard ATRP conditions, with reagents ratio: [I]:[Cu(I)]:[L] 1:2:4, where L is a bidentate ligand: n-propyl-2-pyridyldimethamine and I is CD-initiator. Although polymerisation of MMA was successful with D (dispersity)<1.15, styrene proved to be problematic due to autoinitiation and star-star coupling at high temperatures which did not improve upon decreasing concentration of the initiator. SEC analysis showed that multiarm polymers which have more round shape have lower hydrodynamic radii than their linear...
analogenes due to calibration of SEC with linear standards. Hence, polymers were hydrolysed to cleave the arms and give linear polymers for the SEC analysis to avoid bias. Although this approach provides reliable SEC results, hydrolysis can even take up to a week\textsuperscript{128} and hence slows down the analysis massively. Alternatively, light scattering method can be used.

Chain-chain coupling scales with the number of propagating chain ends and hence is difficult to eliminate during the synthesis of multiarm polymers. It intensifies at high monomer conversion. Hence, to minimise star to star coupling, Reynaud et al. polymerised tert-butyl acrylate in solution rather than bulk with just [1]:[1] molar ratio of [I]:[Cu(I)Br].\textsuperscript{125} Such polymers can be hydrolysed in TFA to give anionic polymers with potential biomedical applications.\textsuperscript{129} It was shown that the same polymerisation conditions can be applied to both α and β-CD bromo/chloropropionyl initiator. Although the obtained dispersity was narrow, detection of linear chains suggests that some leftover 2-bromopropionyl bromide (BPB) from esterification was trapped in the cyclodextrin cavity and initiated growth of unbound chains. This could have been caused by a simplified purification procedure as opposed to chromatographic purification included in similar protocols.\textsuperscript{23,129}

Another study showed that the choice of halogen can have a significant impact on polymerisation. 2-chloropropionate-CD was showed to work better than 2-bromoisobutyrate-CD for ATRP polymerisation of N-isopropylacrylamide (NIPAM) in polar organic solvents, such as acetonitrile.\textsuperscript{18}

CD-initiator was also employed in aqueous ATRP. A cationic monomer - methyl chloride-quaterminated 2-(dimethylamine ethyl) methyl methacrylate was polymerised.\textsuperscript{25} Various results were reported for reactions run at different temperatures due to the effect of temperature on the solubility of the CD-based initiator - 21Br-β-CD initiator was found only moderately soluble in water. At a higher temperature, lower conversions were obtained. To potentially improve polymerisation parameters, fewer initiator moieties per cyclodextrin could be used and remaining hydroxyls could be modified to increase initiator solubility.

SET-LRP, which employs both Cu(I) and Cu(II), was also employed for formation of CD-based polymers.\textsuperscript{109} A pre-activated copper wire (washed with HCl) and wrapped around a stirrer bar was the source of Cu(II). The addition of Cu(II) preserved chain end fidelity and suppressed star coupling. This effect was enhanced by running polymerisation in DMSO - solvent promoting disproportionation. The full conversion was achieved while maintaining living ends which allowed for chain extension to form tri-block-copolymer.

The methods reported so far require relatively big quantities of copper, usually >4000ppm. High content of copper can be problematic, particularly for medical applications, as copper is toxic and difficult to remove completely from the final product.\textsuperscript{130} Matyjaszewski et al. applied simplified eATRP to prepare star block copolymers with a cyclodextrin core with as little as 50ppm of Cu complex.\textsuperscript{24} In eATRP by applying potential, Cu(I)/Cu(II) ratio can be controlled and so the reaction rates and termination. Similarly, ARGET-ATRP was applied for the preparation of (meth)acrylate block copolymers.\textsuperscript{127} ARGET-ATRP is based on continuous regeneration of active metal species – namely reduction of Cu(II) which arises from irreversible radical termination. Additionally, the reducing agent such as ascorbic acid or Sn(EH)\textsubscript{2} can scavange oxygen and other radical inhibitors improving livingness of the polymerisation.

3.1.2 Prepared via ring-opening polymerisation

Cyclodextrin, both native and functionalised, can be also exploited as an initiator for anionic ring-opening polymerisation (ROP). It is well-known that using native cyclodextrin gives ill-defined structures due to the difference in reactivity between three hydroxyl groups.\textsuperscript{131} Further, native cyclodextrin also suffers from poor solubility in common polymerisation solvent. To ensure uniform reactivity, Guégan et al. synthesised 3-hydroxypropyl-CD with 21 and 14 initiating sites.\textsuperscript{132} As polymerisation of ethylene oxide requires strong nucleophile to ensure fast initiation, which is a prerequisite for obtaining living ionic polymerisation, hydroxyl groups were partially deprotonated. Partial deprotonation is sufficient as the fast exchange of labile protons allows all groups to act as initiators. However, due to the poor solubility in DMF, only 20% of hydroxyl groups were deprotonated rather than 30%, the extent that was previously tested and reported successful.\textsuperscript{133} This led to a slow polymerisation, achieving only 18% conversion after 10 days. Although the resulting star-like polymers were well-defined, relatively substantial amount of linear polymer was detected due to some water impurities that could not be removed despite the azetotropic distillation of cyclodextrin. Although ethylene oxide proved difficult to be polymerised from a cyclodextrin core, ROP of ε-caprolactone was reported to proceed fast (95% yield in 3h), giving polymers with narrow dispersity (B<1.15).\textsuperscript{134} Only primary hydroxyl groups were used as initiating moieties and the bottom rim of cyclodextrin was acetylated to reduce hydrogen bonding and improve solubility of the initiator in organic solvents (Figure 11).
Sn(Oct)\(_2\) catalyst ensured fast and simultaneous initiation of all hydroxyl groups. Poly(ε-caprolactone) arms were subsequently coupled with DCC to carboxyl-terminated poly(ethylene glycol) (PEG) to form amphiphilic copolymer that could undergo micellisation.

A judicious choice of functionality of cyclodextrin and monomers allows for growth of two different polymers at the bottom and the top face of the cyclodextrin without protecting groups. Adeli et al. used biocompatible materials to grow poly(lactic acid) arms from primary hydroxyls and poly(2-ethyl-2-oxazoline) from tosylate secondary hydroxyls. The reverse order of polymerisation yielded green solution due to the complexation of Sn(Oct)\(_2\) by nitrogen atoms of poly(oxazoline). The chain end group allowed to tune solubility of the final product. Similarly, Shen et al. showed that ring opening polymerisation and ATRP can be coupled to independently functionalise two faces of cyclodextrin. Such mixed approaches truly leverage the versatility of CD as an initiator. γ-CD proved to be an excellent core for the synthesis of a diblock eight-arm star copolymers via ring-opening metathesis polymerization (ROMP) (Figure 12). As ROMP is functional group-tolerant, hydroxyls not directly involved in the polymerisation did not require any special treatment. Octakis-(6-amino-6-deoxy)-γ-CD was first functionalised with norbornene (Nb) via reaction with N-hydroxysuccinimide precursor (Nb-NHS). After forming an intermediate with Grubbs’ 3rd generation catalyst, cyclodextrin was reacted with norbornene-functionalized hexaethylene glycol. Subsequently, the polymer was coupled with Nb-PEG and quenched with excess ethyl vinyl ether.

Whereas chain extension with hexaethylene glycol was successful, the second chain extension with PEG was less efficient and homo-arm star polymer were observed in SEC. Higher efficiencies were achievable for lower molecular-weight blocks. Nevertheless, multiarm polymers were characterised with narrow dispersity (D = 1.12–1.19).

3.1.3 Prepared via grafting-to approach: coupling reactions

Although ATRP is an excellent tool for core-first synthesis of multiarm polymers, polymerisation of many arms from one core can lead to significant steric hindrance and high local concentration of free radicals. Grafting-to approach can be successful in bypassing star-star coupling and other termination events. Cyclodextrin can also overcome drawbacks of grafting-to approach such as low number of arms. Usually, to ensure efficient conjugation no more than 8 arms are attached to a core but anchoring of even 21 arms to cyclodextrin was successfully reported.

Among coupling reactions, CuAAC coupling between azide and alkyn group is predominant. For more insights and applications of CuAAC coupling to cyclodextrin not only in polymer chemistry, the reader is directed to a detailed review. CuAAC coupling can be easily monitored by \(^1\)H NMR by the appearance of a triazol proton signal around 8ppm or disappearance of azide peak in FT-IR spectrum. Unreacted alkynyl-polymer chains can be removed with the aid of azido-functionalised resin. As described earlier, azido-CD can be prepared via nuclophilic displacement of halogens. This, however, is limited to the primary face of cyclodextrin. To leverage all hydroxyl groups, a hybrid approach is used: the primary face is modified via click reaction whereas the secondary face is used in ATRP or ROP polymerisation. Such heteroarm star copolymers were afforded by coupling alkynyl-PDEA30 (poly(2-(diethy lamino)ethyl methacrylate) to (\(\text{N}^3\))\(_{-}\)-CD-(PNIPAM)\(_{14}\) prepared via ATRP (Figure 13). Alternatively, bromide can be incorporated via esterification with 2-bromopropionic bromide and then subsequently displaced with sodium azide to give β-CD-(\(\text{N}^3\))\(_{21}\) (heptakis[2,3,6-tri-O-(2-azidopropionyl)]-b-cyclodextrin) (Figure 14).

![Figure 12](image12.png) Core-first/graf-from synthesis of a diblock eight-arm star copolymers via ring-opening metathesis polymerization (ROMP) Reprinted with permission from Royal Society of Chemistry, Copyright 2020.

![Figure 13](image13.png) Synthetic scheme for stimuli-responsive double hydrophilic Janus-type ATRP star copolymers, (PDEA)=CD-(PNIPAM)\(_{14}\) based on β-CD derivative via the combination of ATRP and click reaction. Reprinted with permission from Royal Society of Chemistry, Copyright 2009.
Reineke et al. successfully carried out 1,3-dipolar cycloaddition of heptakis-(6-azido-6-deoxy-2,3-di-O-acetyl)-β-CD with alkyne dendrons built of ethyleneamine units. The reaction was catalysed by Cu(I) using copper sulfate/sodium ascorbate and only fully-substituted click clusters were obtained as confirmed by ESI-MS. A different study compared CD-PEG conjugates for biological applications. Different lengths, namely Mn 550, 2000 and 5000, of alkyne-functionalised PEG methyl ester were coupled to heptakis-(6-deoxy-6-azido)-β-CD. All reactions were run at 70°C as below this temperature reactions gave low yields. Although coupling reactions of PEG2000 and PEG5000 gave high yields: 99% and 88% accordingly, mass spectrometry analysis revealed the presence of hexa-substituted product, apart from the desired hepta-substituted one. On the other hand, the coupling of limitation of grafting to approach which becomes less efficient as steric hindrance increases for longer polymer chains. PEG550 was less efficient giving only 55% yield both only hepta-substituted cyclodextrin was observed. Other coupling reactions were also implemented such as thiol-ene base catalysed Michael addition. Thiol-functionalized cyclodextrin was reacted with various monomers and vinyl-terminated polymers prepared via catalytic chain transfer polymerisation (CCTP) in the presence of dimethylphenylphosphine or hexylamine.

### 3.1.4 RAFT

Although RAFT polymerisation allows for preparation of various polymer architectures, it is less attractive for the synthesis of multiarm polymers. Steric hindrance around RAFT agent as well as direct proximity of neighbouring RAFT agents results in an increased number of termination events. RAFT agent can be attached to the CD core via either R or Z group (Figure 15). Attachment via the R groups leads to radical residing on the core and undesired formation of linear macroRAFT agent and star-star coupling. An alternative approach – attachment of RAFT agent via the Z group, results in the radical always residing on the leaving polymer chain. Nevertheless, as conversion increases, polymer arms start shielding the RAFT moiety. The limited access to the RAFT agent makes macroradical prone to termination. This problem is commonly encountered in the synthesis of multiarm polymers. Hence, controlling the steric hindrance is a prevalent requirement for successful star formation and the use of a hyperbranched core can severely limit the conversion of monomer. Even utilisation of just the primary face of cyclodextrin led to increased termination despite the low concentration of radicals. Preparation of glycopolymer showed that targeting short arm lengths can be more successful, nevertheless preparation of block copolymer failed. It was also observed that it took a long time for all arms to get activated. Similar problems with increased termination events have been observed for nitroxide mediated polymerisation from a CD core with seven TEMPO groups. Steric crowding led to higher local concentration of radical chain ends and increased termination.

### 3.2 CD-pendant polymers

Polymers with cyclodextrin side groups can have interesting applications due to the exposure of dangling cyclodextrins that are free to encapsulate small molecules. They can assemble with chains with complementary pendant groups such as adamantane or cholic acid. Pendent cyclodextrins can be further utilised by turning into ATRP initiators via esterification.
Such macro initiators give unusually high grafting densities brush polymers \((\text{Figure 16})\).\(^{144}\) There are two basic approaches to linear polymers with pendant cyclodextrin groups: homo/copolymerisation of a monovinyl CD-based monomer or conjugation of cyclodextrin onto polymer with a complementary functional side chain \((\text{Figure 17})\).

Attempts to directly polymerise CD-based monomers often lead to low molecular weight polymers due to the build-up of steric hindrance.\(^{107,108}\) Perhaps extending the linker (longer amine) could allow for higher polymerisation degrees. The alternative approach is based on postpolymerisation modification \((\text{Table 4})\). Mono-(6-amino-6-deoxy)-\(\beta\)-CD was coupled to poly(acrylic acid) chains with the aid of a peptide coupling agent - benzotriazol-1-yl-oxytripyrrolidinophosphonium hexafluorophosphate \((\text{PyBOP})\), achieving 4% side chain modification.\(^{145}\) Another common approach starts with the synthesis of poly(glycidyl methacrylate). Analogously to the acid group, the epoxy group can be coupled to mono-(6-amino-6-deoxy)-\(\beta\)-CD.\(^{146}\) This approach allows for the synthesis of high molecular weight polymers. One of the shortcomings of postpolymerisation modification is its potential incompatibility with other monomers incorporated into the chain. Incompatibility can be resolved by ‘click chemistry’ which is characterised with high specificity and fidelity in the presence of various functional groups. Copolyesters with pendant alkyne groups were synthesised and subsequently coupled via CuAAC cycladdition with mono-(6-azido-6-deoxy)-\(\beta\)-CD giving \(\beta\)-CD-functionalized copolyester.\(^{147}\) Copolyesters were made via ring-opening polymerisation of propargyl-modified lactones and \(\varepsilon\)-caprolactone in the presence of \(\text{Sn(Oct)}_2\) as a catalyst and adipic acid as co-initiator.

Propargyl-modified lactones were prepared in two steps: the nucleophilic substitution of cyclohexanone with propargyl bromide and Baeyer-Villiger oxidation with an excess of m-chloroperoxy- benzoic acid. To suppress formation of polyrotaxanes during CD-copolyester coupling, polymer chains were capped with bulky 2,2-diphenylethylamine.

Polymers with alkyne side-chain functionality can be also prepared via ATRP. However, polymerisation of propargyl methacrylate resulted in broad dispersity values and cross-linked networks at high monomer conversion.\(^{148}\) This may be ascribed to the coordination of alkyne groups of the monomer to the copper catalyst. The polymerisation of 3-azidopropyl methacrylate, on the other hand, gave well-defined polymers that were coupled to various propargyl-functionalised molecules. Reactions proceeded with nearly full conversion in less than 2h. Further, it was observed that some coupling reactions were more efficient for azido-polymer chains than analogous azido-monomer. This autocatalytic effect can be explained by complexation of copper complex by triazole formed along the polymer backbone.\(^{149}\)

### 3.3 CD-capped polymers

CD-capped polymers are polymers with cyclodextrin chain end group. They can be prepared by polymerisation from a functional initiator bearing one cyclodextrin moiety or by modification of an existing chain end \((\text{Figure 18})\). CD-capped polymers often serve as a host in preparation of polymers such as amphiphilic block-copolymers via host-guest interactions.
Table 4 Strategies for synthesis of CD-pendant polymers.

| CD precursor | Polymer precursor | Preparation of CD-pendant polymer | Ref. |
|--------------|-------------------|-----------------------------------|------|
| Mono-(6-amino-6-deoxy)-β-CD | Poly(acrylic acid) | PyBOP coupling | 145 |
| Mono-(6-amino-6-deoxy)-β-CD | Poly(glycidyl methacrylate) | Amine-epoxy reaction | 146 |
| Mono-(6-azido-6-deoxy)-β-CD | Propargyl-functionalised copolymers | CuAAC cycladdition | 147 |

Analogously to multiarm initiators, esterification with 2-bromoisoobutryl bromide can provide cyclodextrin ATRP initiator with a single initiating moiety.150 This approach, however, is not free from pitfalls such as formation of undesired multisubstituted cyclodextrin. CuAAC coupling of mono-azido-CD with but-3-ynyl-2-bromo-2-methylpropanoate allows for precise synthesis.151 RAFT polymerisation provides another facile method of introducing cyclodextrin as chain end group by using CD-functionalised CTA agent.152 Alternatively, many chain end groups can be converted to alkyne and then coupled to azido-CD.153

Not only linear polymers can be capped with cyclodextrin. Tian et al. polymerised NIPAM from a cyclodextrin core and used it as a macroinitiator in ATRP polymerisation of monovinyl-CD monomer (Figure 19).114 This created a polymer with merged encapsulation properties of cyclodextrin and thermoresponsive behaviour of PNIPAM. Not all the arms chain extended, suggesting either loss of chlorine atoms or low initiation efficiency. The analogous reaction for linear polymers was successful hence it was concluded that the monomer was not reactive enough to overcome the steric hindrance of the multiarm polymer. A single polymer chain can be also grown directly from native cyclodextrin via ROP of lactone and other cyclic esters. If the polymerisation is run in bulk there is no competition from the solvent and monomer – lactone gets trapped in the cyclodextrin cavity. This activates lactone towards polymerisation and leads to a polyester-tethered cyclodextrin.154 As different cyclodextrins have different cavity sizes, their initiating efficiency varies: β-CD worked best for lactone whereas α-CD failed completely. The essential role of complexation was confirmed in a later study by inhibiting polymerisation with adamantane which was forms strong inclusion with cyclodextrin (Figure 20).26
This further study based on solid-state NMR also gave some mechanistic insights into the polymerisation. Namely, it was suggested that it is not the hydroxyl polymer chain end that attacks another lactone but the new monomer. This was confirmed by using mono-2-O-(5-benzylxoypentanoyl)-beta-CD which despite lack of hydroxyl chain end, initiated polymerisation. The monomer, after being activated by formation of a hydrogen bond between lactone C=O and cyclodextrin OH as observed by FT-IR, gets attached to CD to give disubstituted CD. Then the monomer hydroxyl groups attack carbonyl of the polyester chain and get inserted between CD and polymer chain.

In summary, cyclodextrin is compatible with various polymerisation methods, nonetheless, it is mainly found in ATRP protocols. Out of controlled polymerisation techniques, RAFT and nitroxide mediated polymerisation seem to be the least versatile for the preparation of cyclodextrin-polymer conjugates and there is little literature available concerning those two techniques and cyclodextrins. These techniques are particularly inefficient for the preparation of multiarm polymers. This is as cyclodextrin is a relatively small core and steric hindrance leads to increased termination events. Nonetheless, cyclodextrin has been successfully applied as ATRP and ROP initiator, and CD-functionalised monomers were polymerised in controlled manner. Cyclodextrin is particularly promising for the preparation of multiarm polymers as cyclodextrin can overcome common problems such as low arm density and can be applied in the synthesis of miktoarm polymers as well. There are several synthetic strategies towards cyclodextrin-pendant and capped polymers. Such materials can find applications in building more complex structures via self-assembly.

4 Utilisation of CD-polymer conjugates

Since the 1930s when systematic studies on cyclodextrin have started, cyclodextrin has found applications in various fields. As the strongest features of cyclodextrin are encapsulation of small hydrophobic molecules and availability to form non-covalent interactions it is not surprising that cyclodextrin became popular in pharmaceutical and food industry as well as analytical chemistry. Allying cyclodextrin with controlled living polymerisation has also proved to be fruitful. It is outside the scope of this article to comprehensively review all fields that CD-polymer conjugates have been applied in. For a comprehensive review on cyclodextrin applications in drug delivery, cancer therapy and hydrogels the reader is directed elsewhere. Here, we aim to provide some of the pioneering examples as well as recent studies in those fields to show the newcomers the scope of applications of cyclodextrin-polymer conjugates.

4.1 Gene and drug delivery

Cyclodextrin can be asymmetrically functionalised allowing for combining multiple functions. Introduction of azide moieties on the primary rim and alpha-bromopropionate functionalities on the secondary rim led to the construction of gene delivery vector with magnetic resonance imaging contrast (Figure 21). N,N-dimethylaminoethyl methacrylate (DMA) cationic chains grown with ATRP can complex anionic plasmid DNA whereas the azide handle allows for attachment of functionalized gadolinium complex. Self-assembly led to micellar nanoparticles with entrapped DNA – theranostic nanocarriers. Multiarm polymer enhanced relaxation of the magnetic contrast compared to commercial standards. Analogously, drug rather than gene carriers can be designed.

Figure 19 Schematic Illustration of the synthesis routes of Star-PNIPAm and Star-PNIPAm-CD. Reprinted with permission from American Chemical Society, Copyright 2010.

Figure 20 Proposed mechanism of the polymerization of ε-VL by CD in bulk. Reprinted with permission from American Chemical Society, Copyright 2010.
Recently, Becer et al. combined two features of cyclodextrin: encapsulation and selective functionalisation to prepare supramolecular cationic glycosylated polymers (Figure 22). Glycosylation is a common tool for improving bioavailability and targeting properties of such polymers. Cationic polymer arms were grown from the primary face of cyclodextrin leaving access to the hydrophobic cavity from the bottom face for assembly with a glycosylated copolymer containing adamantane. This allowed for better structural optimisation than combining sugar and cationic groups on a single polymer. Increasing length of PDMAEMA improved the transfection efficiency up to an upper limit where increased steric hindrance excluded DNA plasmids. Pendant sugar groups enhanced cellular uptake of DNA in human skin explants and increased expression of delivered DNA compared to non-glycosylated equivalents.

Multiarms polymers with CD core can also self-assemble and encapsulate drugs. pH-responsive statistical PDPA-POEGMA (2-(diisopropylamino)ethyl methacrylate - oligo(ethylene glycol)methylether-methacrylate) copolymers were grown from a cyclodextrin core for chemo/photothermal cancer therapy. Hydrophobic PDPA segment encapsulates hydrophobic doxorubicin (DOX) and releases the drug through swelling in an acidic environment where the segment becomes hydrophilic. Drug release is hence enhanced in the acidic environment of a tumour but leaking to the bloodstream (pH 7.4) is limited. DOX encapsulation efficiency was the highest (40.20%) for copolymer with equimolar content of DPA and OEGMA, providing a balance between water solubility in water and available volume of the hydrophobic domain for drug encapsulation. Co-loading of micelles with photothermal agent enhanced drug activity. However, only 52% of drug was released in in vitro test suggesting micelles might be too stable or drug is bound too strongly.

### 4.2 Functional nanoparticles

Multiarms polymers with a cyclodextrin core can be utilised in preparation of functional nanoparticles. Such polymers can form unimolecular micelles i.e. micelles built of a single molecule. Unimolecular micelles are characterised with increased stability, particularly against dilution. Lin et al. established a synthetic protocol for ferroelectric (PbTiO₃) magnetic (Fe₃O₄) multifunctional nanoparticles. A triblock copolymer CD-P4VPb-PtBA-b-PS was prepared via ATRP from BiBB-functionalised cyclodextrin core and used to direct aggregation of inorganic compounds via interaction of metallic precursor with a specific polymer block (Figure 23). It was found that appropriate solvent mixture was crucial for obtaining well defined, uniform core-shell nanoparticles. This strategy is versatile, allowing for the incorporation of different metals and tuning size by varying lengths of polymer blocks. The external, not bound to metal block can alter the solubility of the particle - PS outer chains aid solubility in common organic solvents and PEO in water. Recently, fluorescent unimolecular micelles were developed for non-invasive optical fluorescence imaging: organelle labelling and tumour localisation.

![Figure 23](image-url)
The ultra-small size of nanoparticles led to enhanced accumulation in tumours. Incorporation of biocompatible CD-core and hydrophilic POEGMA chains bypassed common problems encountered for fluorophore-based materials, namely toxicity and poor biodegradability. Encapsulation of a dye in polymer improved its photostability, compared to stand-alone dyes. The ultra-small size of nanoparticles led to the enhanced accumulation in tumours.

Functional nanoparticles can be also prepared from metallic nanoparticles by attaching cyclodextrin-pendant polymers to the particle surface. Iron oxide particles were coated with silica that provided amino functional end groups that were suitable for further modifications. This allowed for an easy attachment of 2-bromopropionyl bromide that initiated polymerisation of glycidyl methacrylate on the particle surface. Epoxy group of the methacrylate was utilised in appending cyclodextrins. Magnetic nanoparticles were proved to catalyse substrate-selective oxidation of alcohols. They also work as adsorbent, removing impurities such as bisphenol A. Purification was aided by magnetic separation of nanoparticles from clean liquid after complexation (Figure 24).

4.3 Hydrogels

Cyclodextrin is a very promising building block for flexible hydrogels. Although covalently crosslinked hydrogels can be responsive, hydrogels based on supramolecular interactions are more tuneable, responsive to external stimuli such as light or redox stimuli and capable of self-healing. By combining polymer chains bearing CD moieties with polymer chains bearing complementary guest molecules, supramolecular interactions are created that provide durability and improved mechanical properties. Recently, a photoregulated photochromic hydrogel formed of spiropyran functionalized polyacrylamide (PAM) chain, decorated with azobenzene and β-CD pendant groups that could self-heal within 10s was reported. Toughness of the hydrogel was controlled by dynamic host-guest interaction between cyclodextrin and azobenzene as azobenzene isomerises upon irradiation with light (Figure 25). Depending on the wavelength of the light, toughness could be weakened or strengthened. By varying the content of pendant groups on PAM chains, tensile and rupture strain can be tuned, thus allowing for optimisation of mechanical properties of the gel. This is, however, limited by the water solubility of azobenzene functionalised PAM, capping the maximum azobenzene content in the gel.

Easy functionalisation combined with inclusion properties of cyclodextrin allows to prepare covalent/sliding-ring network via one-pot synthesis (Figure 26). Azido-α-CD was used not only as a sliding cross-linker but also as a chain end group to prevent rotaxanes from dissociation. Azido functionality on CD was used to covalently attach dialkyne PEG chains forming the network via CuAAC. Resulting hydrogels are very elastic yet strong. Curing temperature and wt% (weight percentage) of reagents have an impact on the degree of covalency which affects the swelling capability of the network. Further, by precomplexing...
cyclodextrin with sodium 4-(phenylazo)benzoate, almost purely
covalent network can be obtained. The sliding ring networks
were characterised with higher equilibrium swelling ration
values as the dynamic cross-linking points allow the network to
stretch more than static crosslinking points in covalent
analogues. Interestingly, increasing the crosslinking density of
the sliding ring network did not result in a linear trend in
increasing overall stiffness of the material. This can be explained
by the fact that with increasing number of cross-linking points,
the movement of the crosslinks become restricted. Thus, sliding
ring hydrogels cured at high weight % of reagents would behave
similarly to covalently cross-linked networks. Positive
preliminary cytotoxicity revealed potential of such tuneable
hydrogels in tissue engineering applications particularly if
further modified via remaining azido-groups on cyclodextrin to
attach biologically relevant pendants.

Incorporation of cyclodextrin into acrylic hydrogels by
postpolymerisation modification of glycidyl methacrylate led to
preparation of soft contact lenses ocular drug delivery. It was
found that cyclodextrin did not affect majority of hydrogel
physical properties such as swelling degree and viscoelasticity
but decreased friction coefficient. Cyclodextrin was capable of
complexing diclofenac – a nonsteroidal anti-inflammatory drug.
Incorporation of drugs into contact lenses improves drug ocular
bioavailability. Pendant cyclodextrins increased loading
capacity of the hydrogel by 1300%. Attachment of cyclodextrin
to polymer chains slowed down drug release and prevented
leakage of the drug to contact lenses storage solution.
Leveraging hydrophobic cavity of cyclodextrin allows for the
formation of self-healing or thermoresponsive hydrogels.
Dynamic and reversible complexation provides good
mechanical properties while crosslinking prevents complete
dissolution. The choice of a suitable guest is crucial as the
binding affinity will greatly affect the dynamicity of the
hydrogel. Polymers with cholic acid are promising due to
biocompatibility of cholic acid. Such polymers can self-assemble
with polymer chains earing pendant cyclodextrins. Choosing
a guest that can isomerise opens up possibilities of synthesising
photosensitive hydrogels. As photoirradiation leads to
isomerisation of azobenzene connection of alpha-CD bearing
and azobenzene bearing chains is controlled by light. However,
was noted that morphological changes in the
hydrogel structure were relatively difficult to achieve likely due
to the low association constant.

4.4 Miscellaneous applications

Cyclodextrin-centred polymers also found application in surface
science. Multiarm polymers were grafted onto a membrane to
enhance pore selectivity for filtration. Ulbricht et al. showed
that both molecular weight and number of polymer arms,
particularly their flexibility, can influence size-selective ultrafiltration. 7 and 21-arms CD-block copolymers made of
2-dimethylamino(ethyl) and propargyl methacrylates were
‘clicked’ onto azide-modified membrane surface. All polymers
enhanced size selectivity. Less densely grafted stars with longer
arms had the smallest impact on the liquid flux due to their
increased flexibility. This is ascribed to the higher flexibility of
longer arm chains and decreased repulsion between
neighbouring chains. Sample to sample variation was observed
which can likely be attributed to wide dispersity of grafted
polymers (1.2 < D >2.7); potentially optimisation of
polymerisation conditions should yield well-defined polymers
and hence more uniform functionalisation of the membrane.

Multiarm polymers grown from a cyclodextrin core were also
tested as antimicrobial and antiviral agents. High grafting
density and tunability of the number of functional sites allowed
Xiao et al. to optimise structure of a cationic multiarm
polymer. Antimicrobial activity depends primarily on charge
density and hence number and length of polymeric arms as well
as the shape of the polymeric structure. 8-arm acrylamide and
guanidine-based monomer copolymer was found to have high
charge density and a good exposition of charged side chains to
generate strong electrostatic interaction with bacterial
membrane. In addition, incorporation of polyhexamethylene
guanidine hydrochloride made the polymer an effective
antiviral agent.

In summary, cyclodextrin-polymer conjugates found
applications in fields such as nanoparticle preparation,
antimicrobial agents, hydrogels, drug and gene delivery. The
versatility of such conjugates comes from allowing unique
features of cyclodextrin and polymers at the same time. As
cyclodextrin is compatible with controlled polymerisation
methods, the precise engineering of such materials is possible.
Asymmetric functionalisation of cyclodextrin is particularly
promising and resultant hybrid materials are expected to
revolutionise fields such as cancer theranostics which combines
diagnosis and treatment.

5 Conclusions and outlook

Cyclodextrin is a widely available, biodegradable molecule that
allows for the design of novel materials. Two faces of
cyclodextrin can be selectively modified thus enabling the
preparation of asymmetric structures and providing easy access
to multifunctional materials. Cyclodextrin proved to be
compatible with many polymerisation techniques such as
ATRP and ROP but less so with RAFT and NMP. ‘Click chemistry’ can
be used both for grafting polymer arms and attachment of
initiator moiety. Facile reactions such as CuAAC coupling allow
for precise engineering of desired materials. Cyclodextrin-
polymer conjugates have already found applications in drug
delivery, cancer therapy, hydrogels, preparation of inorganic
materials and more.

In order to fully harness the potential of cyclodextrin polymers
more attention has to be paid to the proper characterisation
and purification of prefunctionalised cyclodextrin precursors.
Researchers in the field of polymer chemistry tend to use
protocols from the early stages of the development of
cyclodextrin derivatives. Nevertheless, the synthesis is often not
optimal and in-depth characterisation of the product is missing.
At the same time, organic chemists have been developing new protocols with improved selectivity and yields. Particularly, more attention has to be paid to greener and faster methods such as microwave assisted reactions. There is unrealised potential in asymmetrical functionalisation of cyclodextrins and merging different polymerisation techniques to access materials that are difficult to prepare otherwise. ‘Click chemistry’ has been also mainly limited to CuAAC coupling and other efficient coupling chemistries are waiting to be explored. Advances in the synthesis should lead to the emergence of new applications in the fields that CD-polymer conjugates have already been successfully applied as well as new ones.

6 Conflicts of interest
There are no conflicts to declare.

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