Experimental Study on the Smouldering Combustion of Mineral Wool Insulation in Chimney Penetrations

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Abstract. A large number of fires have been caused by metal chimney penetrations in recent years. One factor that has contributed to these occurrences is smouldering combustion within mineral wool, which is often used as insulation in metal chimneys and their penetrations. Based on reaction to fire performance mineral wool is typically classified as a non-combustible material and in European standards it belongs to category Euroclass A1. However, mineral wool insulation products contain organic material, which smoulders and generates heat at temperatures between 200°C and 500°C. Experimental research indicates that this additional heat may increase the temperature in the chimney penetration materials over 100°C for a limited period of time. The European fire classification of materials does not set requirements to control mineral wool smouldering and limit the subsequent heat release. This experimental study examined the smouldering of mineral wool insulation products specified for chimney penetrations and the heat release generated during the smouldering combustion. The aim of the study was to determine the actual level of temperature increase within the chimney penetration and to estimate the effects of the increase with regard to fire safety. The study also reviewed how the smouldering potential of an insulation material could be considered in the fire classification of products.

Keywords: Smouldering combustion, Mineral wool, Chimney, Fire safety, Heat release

1. Introduction

Despite mandatory certification procedures, some European countries have reported a high number of fires due to the presence of chimneys. For example, in Finland 600–700 building fires caused by fireplaces and chimneys are reported each year [1]. Similar fire safety problems have also been reported in other European countries [2]. In the United States, the majority of residential fires involved solid-fuelled equipment and fires were caused by the ignition of structural frame components [3]. Figure 1 shows a picture from a roof space in which the heat of a metal chimney has led to the ignition of the surrounding roof insulation material and caused a fire. Several reasons for chimney-penetration-induced fires have been
identified. These include, for example, the fact that the actual on-site flue gas temperatures are higher than assumed in chimney design [4, 5], faulty installations, use contrary to operating instructions [6] and the smouldering combustion of mineral wool insulation [6–8]. Previous research by Neri et al. [9–11] and Leppänen et al. [4, 5, 12] have also shown that the standard test set-up and conditions are often different from the actual site installations and may not always lead to a safe solution. All the above issues can increase the temperatures in the roof construction.

Critical details for ensuring fire safety involve the places where chimneys pass through floors, roofs and walls. Based on the product standards of chimneys, the normal flue gas temperatures in residential use vary; for example, for metal chimney products [13], the normal working temperature can be 600°C in the most severe temperature class. Chimney penetration insulation and specific chimney penetration components are used to isolate the hot flue gasses from the flammable insulation materials and structural elements. The purpose of this insulation layer around the chimney is to reduce the surface temperatures adjacent to the chimney down to the level where ignition of the materials does not occur. In chimney product and test standards, the fire safe temperature level at the safe distance is set so low that there is no ignition hazard, even if the material is exposed to this temperature over a protracted period of time. In regulations, a maximum temperature of about 100°C has been considered safe for wood under prolonged exposure [14]. The approval of fire safety for metal chimneys is based on a CE marking and a chimney meets the requirements set in EN 1856-1 [13] if the temperature measured at a safe distance from the outer surface of the chimney does not exceed 85°C, when tested in accordance with EN 1859 [15]. The safe distance to combustible material is specified by the manufacturer of the chimney product. Temperatures inside the area defined by the safe distance are higher and vary between the flue gas temperature and 85°C.

Mineral wool products are often used as the insulation in chimney penetrations. According to standard EN 13501-1 [16], these products are classified as non-com-

**Figure 1.** Chimney roof penetration after a fire.
Mineral wool always contains binder, which is organic. In addition, other organic materials may be used for, for example, dust binding. Earlier research [6–8] have indicated that smouldering combustion of this organic material can occur in the chimney insulation. Smouldering is a flameless combustion that spreads at very low velocity in porous medium and it is characterized by heat release. The initiation of smouldering process is dominated by the oxidation of the medium and can occur when enough energy is available [17]. In case of a chimney penetration insulation, hot flue gasses supply the energy. Stable process occurs if the porous material has sufficient thickness and serves as a heat insulator to prevent heat release from the reaction to the environment [18–20]. In smouldering processes, temperatures ranging between 400°C and 750°C have been measured in the reaction area. Smouldering combustion generates additional heat in the penetration structure that, in turn, increases the temperature of both the penetration insulation and the surrounding floor and roof structures. The additional heat may have a significant effect on the temperature of chimney penetrations and create a potential fire hazard in the surrounding structures. The organic material starts to evaporate when its temperature reaches 200°C and, based on the research by Leppänen et al. [7], all organic material has burned off when the temperature exceeds 500°C. Similar fire performance and pyrolysis of binders have been reported for stone wool in [21].

The amount of additional heat generated by smouldering combustion depends on the amount of organic material and the maximum temperature of the insulation. Figure 2 displays temperatures measured by Inha et al. [6] at the penetration insulation during two successive chimney tests on the same test specimen. In the test specimen, the metal chimney was connected to a sauna stove and installed through a 200-mm thick mineral wool insulation layer. The temperatures in Fig. 2 are measured from a thermocouple that is located in the middle of the mineral wool insulation layer and 100 mm from the face of the chimney flue. The solid curve for Test 1 includes the additional heat release generated by the burning of the organic material. The chimney test was then repeated using the same structure, the result of which is the temperature development depicted by the dashed curve for Test 2. The organic material burned in Test 1, while the additional heat generation was no longer detected in the second test. The difference between the two curves can be interpreted as the additional heat generated by the burning of organic material. The maximum difference in temperature was measured at 140 min and it was 230°C. After reaching its maximum value, the temperature of Test 1 started to decrease, which means that the organic content had burnt off. Temperatures in both tests were then approaching similar temperature level. The estimated duration of the temperature peak in this test was approximately 150 min. Hereafter, the additional heat generated by the burning of organic material is the difference between the maximum values of Tests 1 and 2.

Figure 3 shows an idealised vertical section where a chimney penetrates roof construction. The maximum continuous working flue gas temperature, which in this example is assumed to be 600°C in real use conditions, corresponds to temperature class T600 of EN 1856-1 [13]. The standard test temperature specified for class T600 is 700°C. This is a very typical situation, for example, in Finland.
In this idealized example, a 100-mm-thick penetration insulation is used around the chimney flue and the outer face of the penetration insulation defines the safe distance from the chimney flue. In actual chimney installations the insulation thickness and safety distance required are normally designed on a case-by-case basis and they depend on chimney and penetration construction as well as on properties of penetration insulation products. Roof construction, including thermal insulation and timber roof structures, are installed in contact with the penetration insulation. In the figure, the dashed blue curve represents an allowable temperature distribution where the temperatures do not exceed 85°C outside the safe distance of 100 mm. A solid red curve illustrates the situation where the additional heat generated by the burning of organic material has raised the temperatures. If the temperatures outside the safe distance, area highlighted with red hatch marks in Fig. 3, exceed the ignition temperature of the roof construction materials, the fire hazard is apparent. Smouldering combustion generates heat over a relatively long period of time, which makes the problem even more serious, as the material ignition temperatures decrease when the structures are exposed to high temperatures over a longer period of time.

The building materials used for roofs are usually wood-based, mineral wool insulation and plastic insulation materials. For wood, it is not possible to determine an unequivocal ignition temperature because it is dependent on exposure time. According to Babrauskas [22], wood ignites at around 250°C during short-term fire exposure. After a few hours of exposure, the ignition temperature can be
much lower, below 200°C [14, 22–24]. Wood fibre insulation begins to smoulder at temperatures between 225°C and 290°C [25, 26]. Conservative generic values of the critical ignition temperatures for rigid polyisocyanurate foam (PIR) and rigid phenolic foam (PF) are approximately 300°C and 425°C, respectively [21, 27]. Expanded polystyrene (EPS) melts prior to pyrolysis at around 240°C [21].

Mineral wool insulation products are often used as chimney penetration insulation due to their good fire performance. In the field of fire safety, the most essential characteristics and testing procedures for building products are defined by the harmonized Euroclass system. The system includes a classification system that is based on the reaction-to-fire performance and defines the test methods according to which products shall be categorised. Mineral wool may be included in the highest category Euroclass A1 (non-combustible). The classification and the reaction-to-fire properties are determined either by testing or by the CWFT (classification without further testing) regulation by the European Commission. The EN 13501-1 classification standard [16] sets threshold values for Euroclass A1 products. Non-combustibility is determined in accordance with EN ISO 1182 [28] and calorific value in accordance with EN ISO 1716 [29]. In addition, the method specified in EN 13823 [30] is also used. Commission Decision 96/603/EC [31] contains a list of products that may be included in Euroclass A1 without testing. Mineral wool is one of the materials. None of these materials are allowed to contain more than 1.0% by weight or volume (whichever is the lower amount) of homogeneously distributed organic material. For mineral wool, this is demonstrated by tests carried out in accordance with the method given in EN 13820 [32]. The percentage of organic material in the insulation is called an LOI (Loss On Ignition) value and it states the content of the mass of organic material in relation to the original mass of the sample.
A new testing method for analysing and assessing continuous glowing combustion testing was published in May 2016 and is introduced in EN 16733 [33]. The method, which was developed by the Institut für Brandtechnologie GmbH [34], is based on the method specified in EN 13823 [26]. In the research carried out by the Institut für Brandtechnologie GmbH, three different mineral wool products were tested. The mineral wool binder contents were measured at 3.2%, 4.6% and 3.7% by weight (LOI values). No continuous glowing combustion was detected in the product with the lowest LOI value. This was explained by the low binder content. With the two other mineral wools, glowing combustion was reported. The study report did not include more specific product information or material characteristics.

Previous research have demonstrated that the smouldering combustion of mineral wool insulation products can increase the temperatures of chimney penetration construction during heater operation [6, 7]. However, the research did not consider how sensitive the temperature rise is to the amount of organic content or how high the temperature can rise due to smouldering combustion. For analysing the effects of organic content on the fire safety of chimney penetration construction more information was required on how much the organic content can increase the temperatures and how long is the time the structures are exposed to this additional heat. The above standardized methods are predominantly testing the performance of a building product exposed to an open flame and, therefore, do not correspond to continuous heating conditions. Further research was, therefore, required to determine whether or not the additional heat generated by the burning of organic material should be taken into account when designing chimney penetrations and to develop a robust method for controlling the maximum temperatures in the chimney penetration structures.

In order to answer the above questions an experimental study [8] was carried out at Tampere University (formerly Tampere University of Technology). This paper introduces the test program and the main results of the study and, then analyses and discusses whether or not the additional heat generated by the burning of organic material should be taken into account when designing chimney penetrations, and whether the amount of organic material in penetration insulation should be limited in order to impact heat release and reduce fire hazards.

2. Research Method

The aim of the experimental research was to determine how much additional heat typical mineral wool insulation products used as chimney penetration insulation are generating when exposed to high temperatures. The research consisted of two steps. In the first step, the amount of organic material in each mineral wool test sample was determined by burning the samples in an electric furnace and by comparing the weighed masses of the samples prior and after the burning. In the second step, new samples from the same insulation products as used in step 1 were installed into a support structure placed on the furnace front opening. In these tests, the temperatures of the mineral wool samples were monitored from various
points of the samples and over a longer period of time. Each sample was exposed
to same temperature twice and the additional heat generated by the burning of
organic material was then measured as the difference between the temperatures
during the two heat exposure. By combining the results of the two steps, it was
then possible to interpret the relationship between the organic content and the
maximum temperature rise in a mineral wool sample and further in a product.
The method is presented in Fig. 4.

The details of the test program, equipment and methodology are described in
the following subsections.

2.1. Test Program

The amount of organic material in the mineral wools affects the heat generation
and, therefore, several mineral wool products with different amounts of organic
material were included in the experimental test program. Three different manufac-
turers were included in the study and seven different products were tested. All of
the tested products except one were primarily intended to serve as chimney pene-
tration insulation. The other product was intended for steelwork fire protection.
According to the Declaration of Performances by the manufacturers, all of the
tested products were classified as Euroclass A1 (non-combustible).

This report does not contain the actual product names, but the samples have
been given letter-number combinations. The letters a, b and c refer to the three
manufacturers. The subsequent numbers refer to a particular product from the
manufacturer. The test included four products from manufacturer a. Product a2
was not meant for chimneys. Two products were from manufacturer b and one
product was from manufacturer c. The products were acquired from various hard-
ware stores in Finland.

2.2. Research Equipment

An electric Ceramotherm furnace was used for the study. Its inner dimensions are
550 mm × 700 mm × 800 mm, and the maximum temperature is 1340°C (Fig. 5).

Figure 4. Steps of the experimental research.
The furnace temperature was measured from the centre of the furnace using a type-K sheathed thermocouple. The furnace temperature was controlled with a computer program. Depending on the temperature measured, the computer turned the furnace resistors on and off. When determining the amount of organic material and heat release, 500°C was the highest temperature that was used.

2.3. Organic Content

The amount of dissipated organic material in a mineral wool test sample was determined by keeping the sample in a constant temperature furnace environment over 2 h and measuring the weight loss of the sample. The dissipation of organic material was measured at three different temperatures: 300°C, 400°C and 500°C. The organic material in the mineral wool was burned during the process, with the mass of organic material dissipated at different temperatures determined as the difference between the weighed masses.

The material samples used in the tests were of disc shape with a diameter of 90 mm and thickness of 50 mm. Three test samples were prepared from each insu-
A hole saw was used to remove discs from mineral wool boards. The texture of insulation sample a3 was so soft that a hole saw could not be used. Instead, test samples of 100 mm × 100 mm × 50 mm were cut out of insulation material a3.

The test samples were weighed and then dried at a temperature of 105°C. The dried samples were then weighed and set in the electric furnace (Fig. 6). The furnace was first heated to 300°C and kept at this temperature for 2 h. Then the test samples were weighed. The heating was repeated, first to 400°C and then to 500°C, with the same test samples weighed after both rounds of heating. The test differed from the test specified in standard EN 13820 [32] in that the test samples were heated to three different temperatures, whereas the standard test only uses 500°C. In the standard test of [32], the test sample consists of at least eight smaller specimens from different parts of the insulation product. In this research, the number of specimen was limited to three and, therefore, larger specimens were used to cover local differences in the amount of organic material.

2.4. Additional Heat Generated by the Burning of Organic Material

The test arrangement and the preparation and instrumentation of the samples are based on an earlier study [7]. Three tests were carried out using the electric furnace, with each test consisting of four samples installed into support structure placed on the furnace front opening. During the first two tests, a furnace temperature of 500°C was maintained. During the third test, the temperature was 300°C in order to assess the amount of additional heat at a lower temperature. One test included two separate rounds of heating that were named the first and second heating. During the first heating, the organic material in the insulation burned and caused additional heat. During the second heating, the organic material had already burned and did not have an effect on the temperature variation in the insulation. Thus, the second heating is similar to a situation in which there is no organic material in the insulation. The additional heat generated by the burning
of organic material was then determined as the difference between the temperatures measured during the first and second heating.

The test samples were 200 mm square and 100 mm thick. The earlier study [7] concluded that a 100 mm thick test sample represents the maximum heat release generated by the burning of organic material in the test arrangement used. The selected sample thickness of 100 mm is also close to the thickness of typical chimney penetration insulation. The test samples were made of two approximately 50-mm-thick mineral wool boards placed together back to back. One exception was test sample b2, which was made from ten 10-mm-thick mineral wool boards because that particular mineral wool product was only available with that thickness. The test samples were covered with aluminium foil, except for the side facing away from the furnace, to reduce the airflow inside of them. A foil-covered test sample is depicted in Fig. 7.

The temperature of the test samples was measured from the surface facing the furnace between the aluminium foil and the mineral wool, as well as from various points at 10 mm intervals through the cross-section all the way to the side facing away from the furnace. The structure of the test sample and the temperature measurement points are illustrated in Fig. 8. The surrounding room temperature, $T_{amb}$, was measured from a distance of two meters from the surface facing away from the furnace.

The test samples were installed into 100 mm thick support structure placed on the furnace front opening. The support structure was made of two 50-mm-thick mineral wool boards tied together using threaded rods and nuts, as shown in Fig. 9. Four square-shaped holes were cut into the boards into which the test samples were placed. The side dimensions of the holes and test samples were 200 mm.

Figure 7. Test sample, 200 mm square and 100 mm thick, ready for the test pictured a from the side facing away from the furnace and b from the side facing the furnace.
Two tests were carried out at a furnace temperature of 500°C and one test at 300°C. One test included two separate rounds of heating. At the start of the first heating, the furnace temperature was raised from room temperature to a target

**Figure 8. The cross-section of the 100 mm thick test sample used in the furnace tests and the measurement points.**

**Figure 9. Test samples, 200 mm square, installed into the holes in the support structure from a the side facing away from the furnace and b from the side facing the furnace. Pictures were taken after the test.**
temperature of 500°C. During the heating of the furnace, 50-mm-thick mineral wool slabs were used to cover the holes reserved for the test samples. After the furnace temperature had plateaued at 500°C, the hole covers were removed, one by one, and replaced with the test samples. The test was then continued at 500°C until the temperatures measured from the samples did not change anymore, after which point the furnace was turned off. During the second heating, the test samples from the first heating were tested again in a manner similar to the first heating. The third test including four samples was carried out in a similar fashion, but the furnace temperature was set to 300°C.

In this study, the additional heat generated by the burning of organic material was determined as the difference between the highest temperatures measured during the first and second heating. The highest temperatures were determined based on the measured temperature development at different points of the test sample cross-sections. The highest temperatures of the first heating (T1) were determined as the maximum value of the temperature peak whereas during the second heating, temperatures approached the maximum values (T2) at the end of the test. Temperature difference was then calculated as the difference between temperatures T1 and T2. The approach is demonstrated for test sample b1 in Fig. 12a.

2.5. Measurement Uncertainty

In the tests determining the content of organic material, the main source of uncertainty related to the measurement method is the accuracy of determining the dimensions of the test samples. The softest mineral wools are easily compressed, which made measurement more difficult. A hole saw was used to ensure maximum accuracy for the size of the test samples. For mineral wool a3, a hole saw could not be used, which increased the uncertainty regarding the measurements. Test samples for mineral wool a4 were cut from a cylinder-shaped pipe section, thus increasing the dispersion of the test sample thicknesses.

In the tests determining the heat generated by the burning of organic material, the temperature measured from a test sample is largely dependent on the distance of the measuring point to the heated surface. Even a small deviation in the measurement point location can have a significant impact on the temperature measured. With soft wool insulation in particular, it is difficult to determine the accuracy of the installation. This inaccuracy does not, however, affect the calculation of the temperature rise because the first and second heating are carried out with the same test samples and the measuring points remain the same. The deviations in measuring point locations do have an effect on the estimated shape of temperature distribution, but the effect of the deviations on the results (i.e., the allowed amount of organic material) of this study can be assumed to be small.

One important source of uncertainty in the method presented above is that the amount of organic material (Step 1) and the temperature rise (Step 2) were measured from different test samples taken from the same mineral wool product. As the binder density can vary in an insulation product, it is possible that the organic content used in the two tests was slightly different.
3. Test Results

In this section, the results of the experimental test program described in Sect. 2 are introduced. The organic content of different mineral wool samples and the measured temperature development in the samples are reported in sub-clauses 3.1 and 3.2, respectively. Based on the results, the relationship between the organic content and the corresponding temperature rise can be determined. The results also demonstrate the magnitude and duration of the temperature peak.

3.1. Organic Content

Table 1 shows the amounts of organic material in the test samples used in the furnace tests. The contents of organic material varied between 1.3% and 4.5% by weight, but these contents do not tell us anything about the crucial factor for heat release (i.e., the amount of organic material burned). A better indicator is the amount of organic material by volume. The lowest amount of organic material by volume was in insulation product a3, approximately 1.0 kg/m³. The largest amount of organic material by volume was measured in insulation b1, over 5.0 kg/m³. Overall, it can be stated that the denser mineral wools contained more organic material.

The dissipation of organic material was measured at three different temperatures: 300°C, 400°C and 500°C. A clear mass loss was observed at temperature of 300°C. The most significant portion of organic material had dissipated at 400°C, but a small amount of it was still burning at 500°C. Figure 10 depicts the amount of dissipated organic material by volume in different test samples and at different temperatures.

The results indicated some dispersion for the amount of organic material in mineral wool. Table 1 shows that the consistency and mass of test sample b1:2 differed significantly from the other two test samples from the same product. The deviation was considered so large that test sample b1:2 was excluded from the heat release study that was reported in Sect. 3.2.

3.2. Additional Heat Generated by the Burning of Organic Material

Temperature graphs for the first heating of test samples b1 and c1 are illustrated in Fig. 11. Sample b1 represents a product with high organic content (over 5 kg/m³) and c1 signifies a product with low organic content (approx. 2 kg/m³). The graphs represent temperatures on the test sample cross-sections at 10 mm intervals, during the first heating and with the furnace temperature at 500°C. Additional heat release that stabilizes over the course of 180 min was clearly noticeable. The temperature rise is clearly lower and the organic material burns to completion faster when the concentration is low.

With this test arrangement, the largest temperature rise generated by the burning of organic material was measured at approximately the centre of the test sample. When moving towards the surface facing away from the furnace, the additional heat was significantly reduced, as there was less burning of the organic material and less heat release due to a lower temperature. No burning of organic material occurred.
near the outer surface of the test sample, and the additional heat was able to be transferred from the centre to the surrounding area through the surface. Also convection affects the temperatures on the side facing away from the furnace. This outcome differs from a chimney penetration, as the penetration insulation faces the thermal insulation within the roof. This observation means the additional heat is not able to cool down in the same way as during the furnace test.

Figure 12 illustrates the temperature graphs for the first and second heating for the test samples b1 and c1 measured at a distance of 50 mm from the surface facing the furnace. The heat release generated by the burning of organic material was evident during the first heating, but the second heating of the same test samples did not create any heat release. When comparing the results from the first and

| Test sample | Volume (m³) | Mass (g) | Dry mass (g) | Density (kg/m³) | Mass (g) (300°C) | Mass (g) (400°C) | Mass (g) (500°C) | LOI value (mass%) | Amount of organic content (kg/m³) |
|-------------|------------|----------|--------------|-----------------|------------------|------------------|------------------|-----------------|-------------------------------|
| a1          | 1          | 0.00031  | 42.81        | 42.71           | 140              | 42.37            | 41.75            | 41.68           | 2.41% 3.37                  |
|             | 2          | 0.00030  | 43.64        | 43.54           | 146              | 43.16            | 42.57            | 42.47           | 2.46% 3.58                  |
|             | 3          |          |              |                 |                  |                  |                  |                 |                               |
| a2          | 1          | 0.00031  | 55.69        | 55.56           | 182              | 55.06            | 54.17            | 54.08           | 2.66% 4.85                  |
|             | 2          | 0.00031  | 56.11        | 55.99           | 183              | 55.5             | 54.62            | 54.51           | 2.64% 4.85                  |
|             | 3          | 0.00030  | 55.03        | 54.92           | 184              | 54.42            | 53.58            | 53.47           | 2.64% 4.85                  |
| a3          | 1          | 0.00057  | 39.11        | 39              | 68               | 38.69            | 38.43            | 38.43           | 1.46% 1.00                  |
|             | 2          | 0.00060  | 38.52        | 38.42           | 64               | 38.1             | 37.88            | 37.87           | 1.43% 0.92                  |
|             | 3          | 0.00062  | 41.5         | 41.43           | 67               | 41.13            | 40.92            | 40.88           | 1.33% 0.89                  |
| a4          | 1          | 0.00026  | 22.26        | 22.22           | 85               | 22               | 21.69            | 21.63           | 2.66% 2.26                  |
|             | 2          | 0.00026  | 21.59        | 21.53           | 83               | 21.35            | 21.08            | 21.02           | 2.37% 1.96                  |
|             | 3          | 0.00033  | 25.86        | 25.78           | 78               | 25.55            | 25.18            | 25.12           | 2.56% 2.00                  |
| b1          | 1          | 0.00031  | 50.97        | 50.82           | 163              | 50.29            | 49.35            | 49.18           | 3.23% 5.26                  |
|             | 2          | 0.00031  | 40.85        | 40.71           | 133              | 40.29            | 39.46            | 39.32           | 3.41% 4.55                  |
|             | 3          | 0.00031  | 47.87        | 47.75           | 153              | 47.24            | 46.28            | 46.06           | 3.54% 5.42                  |
| b2          | 1          | 0.00032  | 29.95        | 29.77           | 94               | 29.24            | 28.57            | 28.43           | 4.50% 4.21                  |
|             | 2          | 0.00032  | 32.69        | 32.52           | 102              | 32               | 31.29            | 31.12           | 4.31% 4.40                  |
|             | 3          | 0.00032  | 28.78        | 28.62           | 90               | 28.1             | 27.44            | 27.31           | 4.58% 4.12                  |
| c1          | 1          | 0.00032  | 31.48        | 35.67           | 110              | 35.35            | 35.01            | 34.97           | 1.96% 2.16                  |
|             | 2          | 0.00032  | 33.5         | 33.4            | 105              | 33.11            | 32.77            | 32.73           | 2.01% 2.11                  |
|             | 3          | 0.00032  | 35.8         | 31.4            | 99               | 31.14            | 30.85            | 30.8            | 1.91% 1.89                  |

*Result differs significantly from the other values and was thus excluded from the analysis.
Figure 10. The amount of dissipated organic material at various temperatures.

Figure 11. Temperatures on the test sample cross-section at 10 mm intervals, during the first heating and with the furnace temperature at 500°C (solid thick line): a test sample b1; b test sample c1.
second heating, the temperature rise generated by the burning of organic material is visible. For sample b1, the maximum difference in temperature was measured at 60 min and it was 300°C. The duration of this peak was approximately 150 min (2.5 h). For sample c1, the maximum temperature difference was 100°C at 45 min and the duration was 120 min.

Based on the measured temperature data, the organic material can be presumed to have been burned during the first heating. Tables 2 and 3 present the highest temperatures measured during the first (T1) and second (T2) heating, as well as the temperature increase (T2–T1) at various points within the mineral wool cross-sections. The definitions of the highest temperatures are given in sub-Sect. 2.4. Table 4 presents equivalent measurement results for the third test with the furnace temperature maintained at 300°C. The results demonstrate that none of the test samples exhibited significant heat release during this third test. For test sample b1, higher temperatures were measured during the second heating and it can be assumed that no burning of organic material occurred.

Figures 13 and 14 illustrate the temperature rise measured from different points of the mineral wool cross-section at 500°C. The temperature rise refers to the difference between the highest temperatures measured during the first (T1) and second (T2) heating.

Figure 12. Temperatures at the centre of the test sample cross-section during the first (black curve) and second (grey curve) heating with the furnace temperature at 500°C: a test sample b1; b test sample c1.
4. Discussion

The test results reported in Sect. 3 show that the temperature rise generated by the burning of organic material may increase the temperatures of penetration structures up to the ignition threshold of these materials. It is, therefore, important that the maximum possible temperatures and the risk of ignition in the chim-

| Dist. | $T_1$ | $T_2$ | $T_{1-T2}$ | $T_1$ | $T_2$ | $T_{1-T2}$ | $T_1$ | $T_2$ | $T_{1-T2}$ | $T_1$ | $T_2$ | $T_{1-T2}$ |
|-------|-------|-------|------------|-------|-------|------------|-------|-------|------------|-------|-------|------------|
| 0 mm  | 491   | 457   | 34         | 484   | 464   | 20         | 493   | 456   | 37         | 470   | 455   | 14         |
| 10 mm | 482   | 392   | 90         | 0     | 0     | 0          | 529   | 413   | 116        | 419   | 385   | 35         |
| 20 mm | 465   | 375   | 90         | 447   | 400   | 48         | 545   | 388   | 157        | 400   | 366   | 34         |
| 30 mm | 460   | 352   | 108        | 437   | 379   | 58         | 552   | 350   | 203        | 359   | 321   | 38         |
| 40 mm | 422   | 319   | 103        | 397   | 339   | 58         | 561   | 313   | 247        | 342   | 306   | 36         |
| 50 mm | 404   | 301   | 102        | 388   | 330   | 57         | 566   | 282   | 284        | 317   | 281   | 37         |
| 60 mm | 316   | 226   | 90         | 301   | 247   | 54         | 566   | 248   | 318        | 244   | 214   | 30         |
| 70 mm | 265   | 194   | 71         | 234   | 195   | 39         | 518   | 207   | 311        | 180   | 159   | 21         |
| 80 mm | 184   | 132   | 52         | 175   | 144   | 31         | 398   | 158   | 240        | 153   | 133   | 20         |
| 90 mm | 88    | 74    | 14         | 101   | 84    | 18         | 281   | 102   | 179        | 98    | 89    | 9          |
| 100 mm| 50    | 62    | -12        | 71    | 67    | 4          | 99    | 53    | 46         | 61    | 60    | 1          |

Table 2

Highest temperatures at different points of the test sample cross-sections during the first ($T_1$) and second ($T_2$) heating of test series 1 with the furnace temperature at 500°C. The distance is measured from the surface facing the furnace.

| Test 1 | $a_1$ | $a_4$ | $b_2$ | $c_1$ |
|--------|-------|-------|-------|-------|
| Dist.  | $T_1$ | $T_2$ | $T_{1-T2}$ | $T_1$ | $T_2$ | $T_{1-T2}$ | $T_1$ | $T_2$ | $T_{1-T2}$ |
| 0 mm   | 491   | 457   | 34     | 484   | 464   | 20     | 493   | 456   | 37     |
| 10 mm  | 482   | 392   | 90     | 0     | 0     | 0      | 529   | 413   | 116    |
| 20 mm  | 465   | 375   | 90     | 447   | 400   | 48     | 545   | 388   | 157    |
| 30 mm  | 460   | 352   | 108    | 437   | 379   | 58     | 552   | 350   | 203    |
| 40 mm  | 422   | 319   | 103    | 397   | 339   | 58     | 561   | 313   | 247    |
| 50 mm  | 404   | 301   | 102    | 388   | 330   | 57     | 566   | 282   | 284    |
| 60 mm  | 316   | 226   | 90     | 301   | 247   | 54     | 566   | 248   | 318    |
| 70 mm  | 265   | 194   | 71     | 234   | 195   | 39     | 518   | 207   | 311    |
| 80 mm  | 184   | 132   | 52     | 175   | 144   | 31     | 398   | 158   | 240    |
| 90 mm  | 88    | 74    | 14     | 101   | 84    | 18     | 281   | 102   | 179    |
| 100 mm | 50    | 62    | -12    | 71    | 67    | 4      | 99    | 53    | 46     |

Table 3

Highest temperatures at different points of the test sample cross-sections during the first ($T_1$) and second ($T_2$) heating of test series 2 with the furnace temperature at 500°C. The distance is measured from the surface facing the furnace.

| Test 2 | $a_2$ | $a_3$ | $b_1$ | $c_{1-2}$ |
|--------|-------|-------|-------|-----------|
| Dist.  | $T_1$ | $T_2$ | $T_{1-T2}$ | $T_1$ | $T_2$ | $T_{1-T2}$ | $T_1$ | $T_2$ | $T_{1-T2}$ |
| 0 mm   | 503   | 469   | 34     | 453   | 454   | -2       | 527   | 466   | 61     |
| 10 mm  | 524   | 418   | 106    | 387   | 376   | 11       | 553   | 412   | 141    |
| 20 mm  | 520   | 376   | 144    | 361   | 346   | 15       | 563   | 380   | 183    |
| 30 mm  | 527   | 353   | 174    | 350   | 335   | 14       | 573   | 354   | 219    |
| 40 mm  | 492   | 318   | 174    | 326   | 309   | 16       | 562   | 322   | 240    |
| 50 mm  | 475   | 302   | 173    | 312   | 295   | 17       | 552   | 303   | 248    |
| 60 mm  | 387   | 231   | 157    | 173   | 159   | 14       | 469   | 234   | 235    |
| 70 mm  | 299   | 174   | 125    | 169   | 149   | 20       | 395   | 172   | 223    |
| 80 mm  | 228   | 133   | 95     | 87    | 76    | 11       | 278   | 124   | 154    |
| 90 mm  | 125   | 78    | 47     | 84    | 72    | 11       | 228   | 101   | 127    |
| 100 mm | 76    | 49    | 28     | 70    | 63    | 7        | 81    | 48    | 33     |

4. Discussion

The test results reported in Sect. 3 show that the temperature rise generated by the burning of organic material may increase the temperatures of penetration structures up to the ignition threshold of these materials. It is, therefore, important that the maximum possible temperatures and the risk of ignition in the chim-
ney penetration area can be estimated. In this section, the fire-safe temperature levels are first discussed. The actual floor temperatures are then estimated according to the test results of Sect. 3. Finally, the EN standards specifying the fire resistance properties of penetration insulation materials are reviewed in the light of fire-safe amount of organic material they contain.

Based on the experimental results reported in Sect. 3, combustion burning can raise the temperature levels in a chimney penetration insulation up to $300^\circ C$, which means that it is possible that the ignition temperatures of the combustible roof construction materials at safe distance are exceeded. If $200–250^\circ C$ is considered a critical temperature level for the combustible materials, then the additional heat generated by smouldering combustion should cause an increase in temperature lower than $115–165^\circ C$.

The mineral wool insulation products included in this research contained very different amounts of organic material that varied between 1.0 kg/m$^3$ (sample a3) and 5.0 kg/m$^3$ (sample b1). The heat release at different levels of organic content can be visualised with a graph in which the x-axis represents the amount of organic material (kg/m$^3$) and the y-axis represents the temperature rise in test samples at 500$^\circ C$ furnace temperature. The temperature rise of all the eight test samples are shown as blue points in Fig. 15. The results represent the temperatures measured at 50 mm from the surface facing the test furnace. For the amount of organic material, the average value of the three test samples is used (refer to Table 1). The temperature rise is calculated as the difference between the highest temperatures from the first (T1) and second (T2) heating. Figure 15 shows that the temperature rise is almost directly proportional to the amount of organic

Table 4

| Test 3 | a1 | a2 | b1 | b2 |
|-------|----|----|----|----|
| Dist. | T1 | T2 | T1–T2 | T1 | T2 | T1–T2 | T1 | T2 | T1–T2 | T1 | T2 | T1–T2 |
| 0 mm  | 261 | 263 | –2 | 288 | 284 | 3 | 244 | 246 | –2 | 278 | 274 | 3 |
| 10 mm | 252 | 245 | 7 | 257 | 246 | 10 | 186 | 193 | –7 | 249 | 243 | 6 |
| 20 mm | 223 | 217 | 6 | 234 | 224 | 10 | 164 | 170 | –6 | 229 | 225 | 4 |
| 30 mm | 209 | 202 | 7 | 218 | 209 | 9 | 147 | 154 | –7 | 203 | 198 | 5 |
| 40 mm | 191 | 185 | 6 | 197 | 190 | 8 | 130 | 137 | –7 | 178 | 174 | 4 |
| 50 mm | 177 | 171 | 6 | 186 | 179 | 7 | 120 | 127 | –7 | 161 | 156 | 4 |
| 60 mm | 125 | 119 | 6 | 150 | 145 | 5 | 88 | 94 | –6 | 137 | 137 | 1 |
| 70 mm | 104 | 101 | 3 | 112 | 108 | 4 | 67 | 72 | –5 | 115 | 115 | 0 |
| 80 mm | 72 | 70 | 2 | 87 | 84 | 3 | 53 | 56 | –3 | 87 | 88 | –1 |
| 90 mm | 56 | 55 | 1 | 60 | 57 | 3 | 38 | 38 | 0 | 59 | 59 | 0 |
| 100 mm | 43 | 42 | 1 | 40 | 39 | 0 | 32 | 31 | 1 | 36 | 36 | 0 |

*aTemperatures for test sample b1 remained so low that no burning of organic material occurred. Higher temperatures were measured for the test sample during the second heating.
material in the insulation (kg/m³). The dotted straight line in the figure is the best fit to the data points. The largest deviation between the linear best fit and the tested values was in sample b2, which consisted of ten 10-mm-thick mineral wool layers. It is assumed that the temperature rise was higher because this sample had better oxygen supply than the samples consisting of two 50-mm-thick layers. It can be seen from the figure that an organic content of approximately 3.0 kg/m³ would generate temperature rise critical for the fire safety of a chimney penetration. In this study, four out of seven products had organic content that exceeded 3.0 kg/m³.

When the lowest LOI value of the tests carried out by the Institut für Brandtechnologie GmbH [34] is compared to the LOI values of the products included in this study, it can be seen that only two of the products exceed this LOI value. It would appear, based on this comparison, that EN 16733 [33] does not provide a sufficient method to ensure a fire-safe level of organic material. However, further studies are required to verify this observation.

Based on the above analysis, it can be concluded that the current standards used for assessing and classifying insulation products do not sufficiently take into account the effects of the temperature rise generated by the burning of organic material in continuous heating conditions. Further research is required to achieve

Figure 13. The differences in maximum temperature for the first and second heating at different points of the cross-section for test samples whose organic content was over 2.5 kg/m³ (furnace temperature 500°C).
a robust method for controlling the maximum temperatures in the chimney penetration structures.

In the example used in this research it was assumed that the continuous working flue gas temperature is 600°C, which corresponds to temperature class T600 of EN 1856-1 and, which is a very typical situation in Finland. For lower temperature classes the temperatures within the penetration construction are lower leading also to lower temperature rise due to burning of organic material. This can be seen from the results of test series 3 with the furnace temperature at 300°C.

5. Conclusion

The paper presents experimental research on the smouldering of mineral wool insulation products specified for chimney penetrations and the heat release generated during the smouldering combustion. The following five conclusions can be drawn from the results of this study.

The smouldering combustion of the organic content in typical chimney penetration insulation products generates heat and can increase the temperature in chimney penetration materials by up to 250°C for a limited period of time.

**Figure 14.** The differences in maximum temperature for the first and second heating at different points of the cross-section for test samples whose organic content was under 2.5 kg/m³ (furnace temperature 500°C).
Due to the additional heat release from smouldering, the ignition temperatures of combustible roof construction materials located adjacent to chimney penetration may be exceeded during normal fireplace operation. The experimental results show that the temperatures can be significantly higher than the fire-safe temperature levels given in product and test standards. Therefore, a potential fire hazard exists if the heat release is not limited to fire-safe levels.

The additional heat generated by the burning of organic material is proportional to the amount of organic material in mineral wool (kg/m$^3$). A threshold value is required to limit the organic content in the penetration insulation materials to fire-safe level. Based on the results of this study, an organic content of approximately 3.0 kg/m$^3$ would generate a heat release that can lead to the ignition of combustible roof materials.

The European fire classification of materials does not set any requirements for controlling mineral wool smouldering and limiting the subsequent heat release in continuous heating conditions. Based on the results of this study, testing in accordance with EN ISO 1182 [28], EN ISO 1716 [29], EN 13820 [32] and EN 16733 [33] does not ensure the fire safety of chimney penetration structures. Further research is required to define a fire-safe threshold value and develop a reliable assessment method for organic content. Based on the results of this study, it is

![Figure 15. The effect of the amount of organic material on the temperature rise in the middle of the test samples at 500°C furnace temperature. The points represent determined values for products a1, a2, a3, a4, b1, b2 and c1.](image)
also more relevant to determine the amount of organic material by volume (kg/m$^3$) than by the percentage by weight.

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