GIS-based approach qualitative features of sub-surface water from coastal district in Andhra Pradesh

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Abstract
Quality of water, along Mandal headquarters of a coastal district of Andhra Pradesh, India, was assessed prior to and past rainfall season to determine water’s acceptability for consumption using water quality index (WQI). The present work involved collecting groundwater samples from Mandal headquarters’ locations present in the entire district. All the samples were analysed in a comprehensive way for ten physicochemical parameters, including pH, magnesium, calcium, chloride, sulphate, nitrates, total dissolved solids, fluoride, potassium, and sodium. The geographical information system was used for mapping sampling sites. The coordinates of sample collection areas were recorded employing GPS. Correlation matrices for cation–anion were illustrated and from the qualitative results of the samples. Results indicate that WQI prior and past-rainfall seasons ranged from 37.53 to 312.46; 42.04 to 211.89 in 2016, while for 2017, these were in the range of 25.01 to 137.06; 30.06 to 228.83. The present study’s WQI values indicate that the water samples possess poor quality, and no fluoride contaminations were observed. The analysis suggests appropriate treatment of sub-surface water from the study site before consumption.

Keywords Water quality index · Ground water · Remote sensing · Environmental pollution

Introduction

Among the sources of water available, the source catering to the needs of water for consumption, agricultural activities, and industrial drives in semi-arid and arid zones of several nations is groundwater. Across the globe, nearly 65% caters to consumption, 20% is used for livestock and irrigation, while for industries and mining activities, 15% is used (Taloor et al. 2021; Gautam et al. 2017; Falkenmark and Rockström 2006), about one-third of the population across the globe is dependent on groundwater for consumption (Jakeman et al. 2016; Gautam et al. 2016; Adimalla and Venkatayogi 2018) precisely in developing countries like China and India. Several factors govern groundwater dependence, such as population explosion, enhanced agricultural and industrial activities, inadequate water supply, and other miscellaneous requirements. Owing to the importance and need of this supply, it becomes mandatory to monitor the quality along with the quantity of groundwater (Silva et al. 2021; Sinha et al. 2021; Asano and Cotruvo 2004).

It is understood that the hydrochemistry of groundwater in semi-arid zones is governed by factors such as geology, hydrogeology, precipitation, evaporation, interactions among rock and water, dissolution of minerals, the process of weathering, chemical fertilizers from cultivation, industrial effluents, and several anthropogenic activities (Karunanidhi et al. 2021; Adimalla 2021). Good water quality is essential for the expansion of agricultural and human health (Adimalla and Wu 2019). Many groundwater suitability studies have been carried out across the globe and presented that evaporation and interaction among water and rock are the primary sources for contamination of groundwater precisely in North-eastern Tunisia (Dişli 2017). It was reported that evaporation of groundwater tends to influence
the development of sulphate-type contamination in semi-arid zones of North-west China (Adimalla and Taloor 2020; Ram et al. 2021). Over misuse of groundwater is one of the biggest problems in several parts of the world, among which India is one. It has been reported that groundwater extraction in India annually is the maximum in the world that surpasses China and the USA (Adimalla et al. 2020).

The deterioration of groundwater in India is attributed to improper management strategies for groundwater, mismanagement of wastes, growing population, enhanced industrial and urban development, uncontrolled use of fertilizers, improper dumping of municipal wastes, and dumping yards (Rupal et al. 2012; Palanisamy et al. 2007). The decline of groundwater, specifically in semi-arid and arid zones of India, is attributed to high evaporation rates, erratic rainfall, along over-exploitation; on the other hand, there has been a considerable increase in the call for fresh and potable drinking water leading to water calamity in several parts of the nation (Sakram and Adimalla 2018). Understanding the hydrochemical properties of groundwater helps in rating its appropriateness for various purposes (Adimalla and Wu 2019). Several researchers have conducted studies and have contributed significant information; for instance, multivariate statistical analysis and geochemical modelling conducted by Singh et al. 2017 presented that interaction among rock and water, ion exchange, and weathering were crucial factors responsible for governing the quality of groundwater. A study conducted by Raju et al. 2016 using geostatistical techniques on groundwater quality in hard rock aquifers of Andhra Pradesh, specified that extreme use of pesticides and fertilizers was chiefly responsible for reduced groundwater quality and quantity spatiotemporally. Hydrogeochemical characterization of groundwater quality in Telangana and Tamil Nadu using water quality index and geographic information system (GIS) presented that GIS is an efficient tool for identifying unpolluted and polluted zones (Duraisamy et al. 2019).

Over the decades, the science of geoinformatics has emerged as a significant technology for decision-making in various disciplines (Adimalla and Taloor 2020; Jasrotia et al. 2018). Several researchers have carried out studies towards assessing the spatial distribution of groundwater across the world (Jasrotia and Kumar 2014) and have constructively documented the information using GIS (Jasrotia et al. 2013, Dhar et al. 2017). Several authors have researched groundwater using GIS technology (Jasrotia and Kumar 2014; Cortes et al. 2016) towards accelerating sustainable development (Jasrotia et al. 2019).

Indian coastal aquifers and other parts of the world have reported being the worst affected due to over-exploitation, population explosion, and tremendous industrialization in coastal zones threatening coastal freshwater resources and rendering groundwater non-potable. The presence of major cations and anions in groundwater directly influences its quality. Detailed knowledge of water quality is essential for promoting the effective management of groundwater resources. The water quality index is understood to be one of the utmost efficient tools for communicating information regarding a water body class, people engaged in making policy, and involved residents (Patel et al. 2020). A mathematical equation is utilized to convert data on water quality to a single number (Zotou et al. 2020; Karnena and Saritha 2019; Verma et al. 2020). The dawn of technologies, including satellites and GIS (geographical information system), enabled easy mapping of the sampling sites and generated user-friendly and informative maps precisely for understanding the quality of water (Facchinelli et al. 2001; Zhang et al. 2020; Sinha et al. 2021). The focus of the present work was to determine the quality of water, which was carried out employing the method of weighted arithmetic index proposed by (Jha et al. 2020; Tharani and Abhilash 2021), wherein ten essential characteristics, including physical parameters like pH, total dissolved solids and chemical characteristics including magnesium, calcium, fluoride, chlorides, sulphates, nitrates, potassium and sodium were considered for assessment.

**Methodology**

**Collection of sample and analysis of physicochemical parameters**

Groundwater quality of Srikakulam District was analysed prior and past-rainfall seasons during 2016–2017; water samples were collected from bores varying depths of 25 to 100 m. Thirty-eight Mandal Head Quarters of Srikakulam District in A.P were included from where samples were collected for analysis carried out at Environmental science laboratory of GITAM Institute of Science, GITAM (Deemed to be University), and Visakhapatnam using standard methods given by APHA.

**Sampling stations and samples**

Area of study along with sampling stations is shown in Fig. 1 and is listed in Table 1. Water samples from 152 subsurface sources were grabbed from the 38 sites and were subjected to analysis during 2016 and 2017, prior to and past-rainfall season seasons.

**Parameters selected for potability studies**

For potability studies in relation to standards laid by BIS, the collected samples were subjected to analysis for ten crucial quality parameters, including physical and chemical characterization. Physical characteristics studied were pH, Solids in terms of total dissolved solids, while chemical
Fig. 1 Location map presenting study area

Table 1 Latitude and Longitude of selected sampling zones in Srikakulam district

| S. No (S) | Station              | Samples collected | S. No (S) | Station              | Samples collected |
|-----------|----------------------|-------------------|-----------|----------------------|-------------------|
|           | Lat Long             |                   |           | Lat Long             |                   |
| 1         | Amadalavalasa 18° 24' 53'' | 83°53' 57''       | 20        | Palakonda 18° 36' 13'' | 83°45' 22''       |
| 2         | Bhamini 18° 54' 36''  | 83°49' 11''       | 21        | Palasa 18° 46' 19''  | 85°24' 35''       |
| 3         | Burja 18° 31' 06''   | 83° 48' 9''       | 22        | Pathapatnam 18° 44' 53'' | 84°05' 18''        |
| 4         | Eicherla 18° 16' 54''| 83°49' 28''       | 23        | Polaki 18° 22' 35''  | 84°06' 03''       |
| 5         | Ganguvarisigadam 18° 22' 54'' | 83°40' 53''     | 24        | Ponduru 18° 20' 59'' | 83°45' 26''       |
| 6         | Gara 18° 19' 52''    | 84°30' 28''       | 25        | Rajam 18° 27' 20''  | 83°38' 54''       |
| 7         | Hiramandalam 18° 40' 17'' | 83°57' 01''     | 26        | Ranastalam 18° 12' 07'' | 83°41' 23''      |
| 8         | Ichchapuram 19° 06' 53'' | 84°41' 09''     | 27        | R.Amadalavalasa 18° 33' 03'' | 83°44' 21''       |
| 9         | Jalumuru 18° 30' 41'' | 84°02' 15''       | 28        | Santhabommali 18° 32' 10'' | 84°12' 29''       |
| 10        | Kanchili 18° 58' 50'' | 84°34' 55''       | 29        | Santhakaviti 18° 27' 28'' | 83°46' 50''       |
| 11        | Kaviti 19° 00' 41''   | 84°41' 23''       | 30        | Saravakota 18° 35' 31'' | 84°03' 07''       |
| 12        | Kotabommali 18° 31' 04'' | 84°09' 05''     | 31        | Sarubujjili 18° 31' 13'' | 83°53' 36''       |
| 13        | Kothuru 18° 46' 43''  | 83°52' 53''       | 32        | Seethampeta 18° 41' 33'' | 83°48' 53''       |
| 14        | Lakshminarsupeta 18° 25' 06'' | 84°02' 23''   | 33        | Sompeta 18° 56' 45''  | 85°35' 03''       |
| 15        | Laveru 18° 14' 16''  | 83°41' 48''       | 34        | Srikakulam 18°17' 47'' | 83°53' 04''       |
| 16        | Mandasa 18° 52' 11''  | 84°27'30''       | 35        | Tekkali 18° 36' 19''  | 84°13' 49''       |
| 17        | Meliaputti 18° 46' 28'' | 84°10' 23''     | 36        | Vajrapukothuru 18° 42' 03'' | 84°26' 00''      |
| 18        | Nandigam 18° 21' 55'' | 84°08' 32''       | 37        | Vangara 18° 37' 04''  | 83°36' 28''       |
| 19        | Narasannape 18° 24' 56'' | 84°02' 41''     | 38        | Veeraghattam 18°41' 18'' | 83°36' 30''       |
characteristics evaluated comprised of total hardness including calcium and magnesium, fluoride, chlorides, nitrates, sulphates, potassium and sodium. Analysis followed standard protocols laid by APHA, 2015.

**Given below is the methodology applied towards calculating WQI**

**a. Weightage factor** For calculation, first, the weightage of the individual parameter is calculated. Weightage for different characteristics is allocated which is contrarywise relational to suggested standards for interrelated parameters. Therefore,

\[ W_i = \frac{K}{S_i} \]  

**b. Water quality rating** The calculation of rating is based on the equation given below, where \( V_a \) and \( V_i \) correspond to real and perfect values of water quality. Further, for all parameters, the perfect value is 0 excluding pH and dissolved oxygen.

\[ q_i = \left\{ \left\{ \left( \frac{V_a - V_i}{S_i - V_i} \right) \right\}^2 \right\} \cdot 100 \]  

**c. Calculation** Principally, WQI is an assembly of several parameters which can be employed to assess complete water quality (Table 2). Parameters included in WQI are pH, chlorides, calcium, sodium, potassium, magnesium, fluorides, nitrates, sulphates, and total dissolved solids. The numerical value pertaining to rating of the quality is then multiplied by a weightage factor (Table 3) relates to the test’s significance of water quality. Sum of the values that result are then added towards arriving at a complete water quality index.

**d. Water quality index** WQI = \( \Sigma q_i w_i \) (Water quality rating)

\[ q_i = \left\{ \left\{ \left( \frac{V_a - V_i}{S_i - V_i} \right) \right\}^2 \right\} \cdot 100 \]

\( v_a \) = actual value present in the water sample.

\( v_i \) = ideal value (0 for all parameters except PH and DO).

\( W_i = K/S_i \)

**Table 2** Water Quality Scale concerning WQI

| Quality of water   | Water quality index (WQI) |
|-------------------|---------------------------|
| Exceptional       | 0 to 24                   |
| Good quality      | 25 to 50                  |
| Inferior          | 50 to 75                  |
| Bad               | 75 to 100                 |
| Not fit for drinking | > 100                    |

**Table 3** Component wise weightage of characteristics as per BIS

| Characteristic          | Standard value (Sn & Si) | Assigned weightage factor (Wi) |
|-------------------------|--------------------------|-------------------------------|
| pH                      | 8.5                      | 0.1363529                     |
| Calcium                 | 75                       | 0.0154533                     |
| Chlorides               | 250                      | 0.0046360                     |
| Fluorides               | 1.5                      | 0.7726666                     |
| Magnesium               | 30                       | 0.0386333                     |
| Nitrates                | 45                       | 0.0257555                     |
| Total Dissolved Solids  | 500                      | 0.00231800                    |
| Sulphates               | 250                      | 0.004                         |
| Sodium                  | 200                      | 0.005                         |
| Potassium               | 10                       | 0.1                           |

\* All values except for pH are expressed in mg/l

Where \( w_i \) (unit weight). \( k \) (constant) = \( 1/(1/V_{s1} + 1/V_{s2} + \cdots + 1/V_{sn}) \); \( Sn \) = standard value.

**Water quality indices**

Since their formulation, water quality indices were developed by several researchers over time and adopted for various studies. These are understood to be an efficient tool for converting an extensive set of data into a precise, informative, and meaningful form to extract characteristics of the sample’s underlying facts. Weighted Arithmetic Index has been implemented to assess existing water quality status and identify those physicochemical parameters that cause pollution. The present study adopted the WQI given by (Verma et al. 2020; Zhang et al. 2020; Karnena and Saritha 2019; Jha et al. 2020).

**Statistical methods**

Values of Correlation coefficient \( (r) \) were found out employing the correlation matrix for identification of greatly interrelated, correlated parameters of water quality.

**Results and discussion**

Srikakulam District is situated within 18.6177° N and 84.0167° E of Northern latitude and 83°50’ and 84°50’ of Eastern longitude. Srikakulam is the second-most backward district in Andhra Pradesh, India. The community has an industry called Dr. Reddy’s Laboratories at Pydibhimavaram. The district has a lot of scope for development in the future with the forthcoming establishment of a thermal power plant at Sompeta and a nuclear power plant at
Ranasthalam. The district is bordered to a distance by Vamsadhara, Kandivalasagedda, and Bahuda at specific spans of their courses white a line of elevations of the great Eastern Ghats that run from North-East. The study is undertaken for thirty-eight sampling stations throughout the Srikakulam District. Groundwater suitability at diverse locations from study area for consumption and irrigation was evaluated.

**Potability studies (as per BIS)**

The present study assessed groundwater potability based on analysis of physicochemical parameters and assessment of groundwater quality indices.

**Physicochemical analysis**

Based on the physicochemical analysis, the estimation of groundwater quality for its appropriateness towards consumption is one of the present study’s objectives. The graphs and thematic maps below illustrate the result of analysis of samples collected prior to and past-rainfall season seasons of 2016 and 2017. In the study period, the temporal and spatial disparity in each water quality was presented.

**Seasonal variation for the year 2016 and 2017** Seasonal variations of the groundwater's physicochemical characteristics by comparing prior to and past-rainfall season data in the area of study for 2016 and 2017 are as follows.

a) **pH:** During 2016, 6.5 to 8.8 of pH was noted prior to-rainfall season, while past-rainfall season recorded 7.2 to 8.9. The comparison of pH between prior to and past-rainfall season was plotted in the graph as shown in Fig. 2a. An alkaline nature was illustrated by maximum samples from the study. In prior to-rainfall season, 58% of the sample station's pH values are more than the highest desirable level (6.5–8.5) during both the years. The pH range is between 7.34 and 8.93 for prior to-rainfall season and 7.58 and 8.69 for past-rainfall season during 2017.

The thematic maps have been generated by ARC GIS 10.3 software using a spatial analysis tool. The classification of water quality parameter pH is alienated into three categories according to BIS. These are most suitable for drinking (<250 mg/l) and not ideal for drinking (250–600 mg/l) and greater than 600 mg/l. Figures 3a and b and 4a and b show the spatial distribution of water quality parameter pH.

b) **Total dissolved solids (TDS):** For 2016, 79.36 to 2112 mg/l of TDS was recorded during prior to-rainfall season, and 118.02 to 1296 mg/l was observed in the past-rainfall season, indicating maximum sampling sites in the study falling above standard level. Thus, it can be understood that agricultural activity was the chief anthropogenic impact leading to disparity in spatial and temporal run-off of the location. Figure 2b shows the total dissolved solids graph compared between prior to and past-rainfall season. During 2017, 70.4 to 3584 mg/l of TDS was reported during prior to-rainfall season and 134.4 to 3392 mg/l for past-rainfall season, indicating that maximum sampling sites fall above standard levels.

These are most suitable for less than desirable value for drinking (<500 mg/l) and allowable for consumption (in between 500 and 1500 mg/l). Figures 3c and d and 4c and d show the spatial distribution of TDS during both prior to and past-rainfall season.

c) **Chlorides:** The chloride value was in the range of 20 to 471 mg/l in prior to-rainfall season, and 30 to 500 mg/l in past-rainfall season was observed during 2016. Figure 2c shows the fluoride graph compared between prior to and past-rainfall season. The concentration of chloride in most of the samples was more significant than the highest desirable level (250 mg/l). For 2017, the chloride value ranged from 14.184 to 496.44 mg/l for prior to-rainfall season, and 30 to 1360 mg/l for past-rainfall season was observed. The chloride concentration in most of the samples was more significant than the maximum enticing level (250 mg/l) prescribed by BIS.

Nevertheless, values from the study are found to be entirely lower than maximum permissible limit (1000 mg/l). Anthropogenic sources of chloride are from bleaching agents by communities nearby bore well and septic tank effluents. The classification of water quality parameter chloride is alienated into three categories according to BIS. These are most suitable for drinking (<250 mg/l) and not ideal for drinking (in between 250 and 600 mg/l and greater than 600 mg/l). Figures 3e and f and 4e and f show the spatial distribution of water quality parameter chlorides in both prior to and past-rainfall season.

d) **Fluorides:** During 2016, subsurface water generally contains fluoride, which is dissolved from the geological formation. Figure 2d shows the fluoride graph compared between prior to and past-rainfall season. The concentration of Fluoride in sampling stations ranged between 0.01 and 1.2 mg/l in prior to-rainfall season and 0.1 and 1.3 mg/l in past-rainfall season seasons. Enhanced fluoride concentrations might lead to deformities like dental and skeletal fluorosis. For 2017, concentration of Fluoride ranged between 0.02 and 0.88 mg/l in prior to-rainfall season and 0.02 and 0.89 mg/l in past rainfall season seasons. The classification of water quality parameter fluoride is alienated into three categories. These are most suitable for less than desirable value for drinking (<1 mg/l) and not suitable for drinking (in between 1 and 1.5 mg/l and more significant than 1.5 mg/l). Figu-
Fig. 2  Depicts the comparisons of elemental concentrations of pre- and post-monsoon

(a) pH comparison in prior to and past rainfall season.

(b) Comparison of TDS prior to and past rainfall season.

(c) Concentration of chloride prior to and past rainfall season.
(d) Comparison of fluoride concentration in prior to and past rainfall season.

(e) Comparison of nitrates concentration in prior to and past rainfall season

(f) Concentration of sulphates prior to and past rainfall season
(g) Comparison of sodium amount in prior to and past rainfall season

(h) Concentration of potassium prior to and past rainfall season.

(i) Comparison of calcium concentration in prior to and past rainfall season.
Figure 3 (g and h) and Figure 4 (g and h) show the spatial distribution of water quality parameter fluorides in both prior to and past-rainfall season.

e) **Nitrate:** During 2016, nitrate levels from current study stations ranged from 0.62 to 409.2 mg/l during prior to-rainfall season, while 0.03 to 51 mg/l for the past-rainfall season. Figure 2e shows a graph plotted for the comparison of prior to and past-rainfall season for nitrates. Maximum stations are in the enticing limit of 45 mg/l—nitrate levels, since more than 45 mg/l causes Methemoglobinemia in infants. For 2017, the present study's nitrate levels ranged from −1.82 to 49.6 mg/l in the prior to-rainfall season and −1.45 to 71.45 mg/l in the past-rainfall season. Most sampling sites fall well in the enticing limit of 45 mg/l.

Classification of nitrates is alienated into three categories according to BIS. These are most suitable for less than desirable value for drinking (<45 mg/l) and unsuitable for drinking is greater than 45 mg/l. Figures 3i and j and 4i and j show the spatial distribution of nitrates in both prior to and past-rainfall season.

f) **Sulphate:** During 2016, sulphate concentrations in the sampling sites ranged from 5 to 278 mg/l in prior to-rainfall season and 0.82 to 374 mg/l for past-rainfall season. Figure 2f shows a graph plotted for the sulphate's comparison between the prior to and past-rainfall season. For 2017, sulphate concentrations from sampling sites ranged from 4 to 270 mg/l in prior to-rainfall season and 6 to 213 mg/l in past-rainfall season, respectively. The classification of water quality parameter sulphates is alienated into two categories according to BIS. These are suitable for drinking (<250 mg/l) and not fit for consumption (>200 mg/l). Figure 3k and l and 4k and l show the spatial distribution of sulphates in both prior to and past-rainfall season.

g) **Sodium:** During 2016, sodium concentration in sampling sites ranged from 12.19 to 736 mg/l in prior to-rainfall season and 6.9 to 215 mg/l in past-rainfall season, respectively. Figure 2g shows a graph plotted for the sodium comparison between the prior to and past-rainfall season. For 2017, sodium concentration in sampling sites extended from 0.78 to 98.8 mg/l in prior to-rainfall season and 19.1 to 804.2 mg/l in past-rainfall season, respectively. The classification of water quality parameter sodium is alienated into two categories according to BIS. These are suitable for drinking (<20 mg/l) and not ideal for drinking (>200 mg/l). Figure 3m and n and 4m and n show the spatial distribution of sodium in both prior to and past-rainfall season.

h) **Potassium:** During 2016, potassium concentration in sampling sites ranging from 0.69 to 206.7 mg/l in prior to-rainfall season and 0.39 to 160 mg/l in past-rainfall season, respectively. Figures 2h shows a graph plotted for the potassium comparison between the prior to and past-rainfall season. For 2017, amount of potassium in sampling sites ranged from 0.3 to 147.8 mg/l in prior to rainfall season and 0.53 to 260.9 mg/l in past rainfall season. The classification of water quality parameter potassium is alienated into three categories according to BIS. These are suitable for drinking (<100 mg/l) and not fit for consumption (>200 mg/l), and not acceptable (>500 mg/l). Figure 3o and p and 4o and p show the spatial distribution of calcium in both prior to and past-rainfall season.

i) **Calcium:** During 2016, calcium concentrations among the sampling stations ranged between 10 and 172 mg/l for prior to-rainfall season and 8 and 144 mg/l for past-rainfall season. In some of the stations, the concentrations fell above the prescribed standard of 75 mg/l dur-
ing both the years. Figure 2i shows a graph plotted for the calcium presents comparison among the prior to and past-rainfall season. The enhanced values are owed to the abundant availability of limestone in the location resulting in enhanced solubility of calcium ions. For 2017, amount of calcium from current study area ranged between 8 and 68 mg/l for prior to-rainfall season and 16 and 224 mg/l for past-rainfall season. Classification of calcium is alienated into three categories according to BIS. These are suitable for drinking (<75) and not ideal for drinking (75–200 and >200). Figures 3q and r and 4q and r show the spatial distribution of calcium in both prior to and past-rainfall season.

j) **Magnesium**: During 2016, magnesium in the current study ranged from 47 to 464 mg/l in the pre-rainfall season and 39 to 700 mg/l in the past-rainfall season. Magnesium falls above the standard limit in some of the sampling stations for both seasons. The magnesium concentration may be very high credited to the dissolution of magnesium, gypsum, dolomite and calcite. Figure 2j shows a graph plotted for the magnesium comparison between the prior to and past-rainfall season. For 2017, magnesium in sampling sites ranged from –7.2 to

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**Fig. 3** Thematic maps of the different elements of pre- and post-monsoon of 2016
90 mg/l for prior to-rainfall season and 4.86 to 260 mg/l for the past-rainfall season. Few sampling sites reported magnesium concentrations above the standard desirable limit for both seasons. The classification of water quality parameter Magnesium is alienated into three categories according to BIS. These are most suitable for less than desirable value for drinking (< 30 mg/l) and permissible for drinking (in between 30 and 100 mg/l) and unsuitable for drinking is more significant than 100 mg/l. Figures 3s and t and 4s and t show the spatial distribution of magnesium in both prior to and past-rainfall season.

**Water quality indices**

Adequate quantity and acceptable quality are essential parameters of water required for the sustenance of life. To appraise groundwater’s suitability from current study area towards consumption, evaluation of water quality index using a weighted arithmetic index method has been carried out as per the methodology mentioned in Sect. (2.1.4). The status of water quality has been studied for thirty-eight Mandal headquarters of Srikakulam District.

**Variation of WQI seasonally** Low values of WQI imply that water does not have any pollutants arising from sampling...
Fig. 5  Thematic Map of WQI for Prior to and Past Rainfall season in 2016

Fig. 6  Thematic Map of WQI for Prior to and Past-Rainfall season in 2017
When WQI is significantly over 100, it is found that water is polluted and unfit for consumption. WQI values for prior to and past-rainfall seasons are illustrated in Figs. 5 and 6. Figure 5 is developed to indicate percentage disparities of diverse traits of water prior to and past-rainfall seasons. The study shows that the outcomes of the WQI values in prior to and past-rainfall seasons range from 37.53 to 312.46, 42.04 to 211.89, respectively, for the year 2016. Figure 6 illustrates the variations of other water attributes in prior to and past-rainfall seasons of 2017. WQI values prior to, past-rainfall seasons are in the range of 25.01 to 137.06, 30.06 to 228.83, respectively, for the year 2017. Using Microsoft Excel bar diagrams, Figs. 7 and 8 are generated to indicate the disparities of diverse water quality in prior to and past-rainfall seasons.

Correlation coefficient (CC) matrix of water parameters

CC of nearly 1 or −1 lies between −1 and +1. Further, CC falling around zero implies none relationship, while positive relationship is indicated by positive values, and an inverse relationship is indicated by negative values of r. During prior to-Rainfall season (2016), Electric conductivity (E.C.) showed robust optimistic and significant correlation with Na+, TDS, Cl− and showed a negative relation with magnesium. Magnesium showed a negative relationship with almost all parameters. All the remaining parameters showed a weak relationship with one another. In the year 2017 prior to-rainfall seasons, EC showed a strong relationship with TDS and Sulphate, whereas the remaining parameters showed an average relationship, while during the past-rainfall season, EC showed a strong
relationship with $\text{SO}_4$, $\text{Cl}$- and TDS, which led to strong relationships, whereas all remaining parameters showed weak relationships.

**Conclusions**

- From the analysis of prior to and past-rainfall season data and the average values obtained, it has been inferred that specific parameters like Calcium, Magnesium, TDS have increased. Remaining parameters like $\text{pH}$, Chlorides, Fluorides, Nitrates showed a decrease in concentration, which is attributed to an increase in concentration due to more significant leaching and a reduction in concentration due to dilution.
- The spatial representation of quality of groundwater from study area is presented as final output, which indicates that subsurface water from the current study area requires some degree of treatment to be fit for consumption. Data from the present study enable understanding water quality and developing appropriate management steps to protect subsurface water sources.
- No fluoride-contaminated water was observed in the current study. Out of thirty-eight groundwater stations in the study area, ten stations show nitrate concentration more than 45 mg/l causing blue baby syndrome in infants. It is suggested to choose an alternative source of water supply.
- The method of Weighted Arithmetic Index was adopted for assessing the water quality status. WQI values of subsurface water samples for prior to and past-rainfall season seasons illustrate a slight alteration that are not noteworthy concerning potability and groundwater quality. Trends along long-term for overall index values are noteworthy concerning potability and groundwater quality variables.
- As per the WQI scale, the selected stations from where groundwater was collected is classified on a scale from excellent to very poor and, in certain areas the quality of water was not fit for drinking. The highest value of WQI (312) is observed at S20 in Prior to-rainfall season (2016) and the minimum value of WQI (25.87) at S28 past-rainfall season.

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**Declarations**

**Conflict of interests** The authors declare that they have no competing interests.

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