Regular Article

Synthesis of One Double Bond-Inserted Retinal Analogs and Their Binding Experiments with Opsins: Preparation of Novel Red-Shifted Channelrhodopsin Variants

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In optogenetics, red-shifted channelrhodopsins (ChRs) are eagerly sought. We prepared six kinds of new chromophores with one double bond inserted into the polyene side chain of retinal (A1) or 3,4-didehydroretinal (A2), and examined their binding efficiency with opsins (ReaChR and ChrimsonR). All analogs bound with opsins to afford new ChRs. Among them, A2-10ex (an extra double bond is inserted at the C10–C11 position of A2) showed the greatest red-shift in the absorption spectrum of ChrimsonR, with a maximum absorbance at 654 nm (67 nm red-shifted from that of A1-ChrimsonR). Moreover, a long-wavelength spectral boundary of A2-10ex-ChrimsonR was extended to 756 nm, which reached into the far-red region (710–850 nm).

Key words channelrhodopsin; retinal analog; chromophore; red-shift; optogenetics

Introduction

Channelrhodopsin-1 (ChR1, 1) was discovered from Chlamydomonas reinhardtii by Nagel et al. in 2002. 1 It is a member of the microbial rhodopsin and is a light-gated cation channel. ChR1 consists of the chromophore (all-trans-retinal, 2a) and a seven-transmembrane helix apoprotein (opsin). The retinal covalently binds to the ε-amino group of lysine residue 296 (K296) through a protonated Schiff base (PSB). Upon absorption of light, photoisomerization of retinal from the all-trans to the 13-cis forms occurs with a conformational structure change of the opsin, and then ChR1 activate the membrane with opening the channel and ion transportation (depolarization). Since the first finding of ChR1, plenty of similar channelrhodopsins (ChRs) have been discovered from archaea, bacteria and eukaryotes.7

Recently, a new technique for treatment of cellular perturbation tools called optogenetics has been developed.3–6 Optogenetics is a technology for controlling specific neural activities by light irradiation that arises from the depolarization of ChRs-expressing neural cells. This technique is less invasive compared to the traditional electrode method that affects other cells. Therefore, optogenetics are deemed to be a useful tool to clarify the neural circuits and the mechanism of various neural diseases.3,8 ChRs now used in optogenetics are mostly sensitive to blue-green light (430–550 nm) and bear several limitations, such as low expression levels, ability of depolarization, kinetics (response of excitation and desensitization), spectra response and ion selectivity due to their biological and physical properties. To overcome these problems, several approaches for the development of new ChRs have been achieved.9 Among them, to create the red-shifted ChRs that are sensitive to long-wavelength light is highly attracted because the red-shifted ChRs can not only provide new optogenetic tools, which were controlled by currently unused wave-length light, but also deeply penetrate the targeted neural tissues without the influence of other tissue components such as water and blood (scattering and absorption light).

One method to create red-shifted ChR is to introduce mutations of the amino acids of opsin and several such achievements have been reported.10–12 Another approach has been to replace retinal with another red-shifted chromophore having a system of extended conjugation.13–15 In connection with our study on development of new red-shifted ChRs,16 very recently we designed and prepared new chromophores in which a one double bond was inserted into the polyene side chain of all-trans-retinal (A1, 2a) and 3,4-didehydroretinal (A2, 2b) at C6–C7, C10–C11 or C14–C15 positions (nomenclature based on retinal numbering), which were named A1-6ex (3a), A2-6ex (3b), A1-10ex (4a), A2-10ex (4b), A1-14ex (5a) and A2-14ex (5b) (Fig. 1). All analogs except for 3a were incorporated into CIC2, which is one of the most widely used ChR variants in optogenetics (an absorption maximum at 477 nm),17 and among the newly created ChRs, A2-10ex-CIC2 showed the longest absorption maximum at 510 nm.18 These results indicated that polyene chain-elongated retinals could be incorporated into the retinal-binding pocket of CIC2. However, the red-shift value of absorption maximum of A2-10ex-CIC2 was not enough compared to the ideal red-shifted ChRs, whose absorption maxima are required in the near-infrared region (750–1500 nm). We anticipated that it might be possible to produce the even more red-shifted new ChRs by changing opsins for binding. As other opsins, we chose ReaChR19 and ChrimsonR20 because their absorption maxima were 61 nm and 110 nm red-shifted, respectively, compared to that of CIC2. Although the data of A2 analogs (3b, 4b and 5b) were already reported in the literature,20 we describe here a full ac-
count of the synthesis of six kinds of chromophores and their binding experiments with other opsins.

Chemistry

With the exception of 5a,21) our designed one double bond elongated retinal analogs 3–5 had been unknown. We therefore planned a systematic synthetic route for 3–5 from the known aldehydes: a 3-step sequence of Horner–Wadsworth–Emmons reaction/reduction/oxidation (Chart 1). The first step of the Horner–Wadsworth–Emmons reaction with aldehyde using triethyl phosphonoacetate (C2-phosphonate) or triethyl 3-methyl-4-phosphonocrotonate (C5-phosphonate) would afford the corresponding C2- or C5-elongated conjugate esters in a highly E-stereoselective manner. In the second reduction step, the resultant conjugate ester would be converted to the allyl alcohol by using lithium aluminum hydride (LiAlH₄) or diisobutylaluminum hydride (DIBAL). The third oxidation step from allyl alcohol to conjugate aldehyde would be achieved by using manganese dioxide (MnO₂). If we appropriately chose C2- or C5-phosphonate in this 3-step sequence, repeating this 3-step sequence would ultimately provide access to the desired retinal analogs 3–5. This methodology contains a quite traditional polyene construction, but we believed this protocol would be a practical, scalable, and robust procedure.

A1-6ex (3a) was prepared from commercially available β-cyclocitral (6a) by a C2-/C5-/C5-elongated sequence (Chart 2). In the Horner–Wadsworth–Emmons reaction of C2-phosphonate, the E-isomer was obtained exclusively. On the contrary, in the case of C5-phoshonate, the newly produced conjugate ester was obtained as a stereoisomeric mixture with a terminal double bond (E:Z = 9:1 to 18:1), and it is well known that N,N-dimethylpropyleneurea (DMPU) plays an important role for this high stereoselectivity.22) As the separation of stereoisomers was difficult, the stereoisomeric mixture was used for the next reaction, and the all-E isomer was easily isolated after conversion to the aldehyde by column chromatography. In a similar fashion, A2-6ex (3b) was prepared from safranal (6b).

A1-10ex (4a) and A2-10ex (4b) were prepared from β-ionylideneacetaldehyde (9a)23) and 3,4-didehydro-β-ionylideneacetaldehyde (9b)24) by C2-/C5-elongated sequence (Chart 3).

![Chart 1. Three-Step Sequence for Conjugate Aldehyde Construction](image)

![Chart 2. Synthesis of A1-6ex (3a) and A2-6ex (3b)](image)
Although A1-14ex (5a) was a known compound as 14-apo-β-carotenal, which was produced by the oxidative degradation of ββ-carotene,\textsuperscript{21} we prepared 5a by the C2-elongated sequence from retinal (2a).\textsuperscript{25} In a similar fashion, A2-14ex (5b) was prepared in 54% overall yield from 3,4-didehydroretinal (2b)\textsuperscript{20} (Chart 4).

**Binding Experiment with Opsins**

With the new chromophores in hand, we next examined their binding efficiency with opsins of ReaChR and ChrimsonR. All analogs bound to both opsins to form new ChRs. The spectra from ReaChR and ChrimsonR are shown in Fig. 2 and 3, respectively, where the absorption spectra were normalized to absorbance 1.0 at 278 nm. These data were summarized in Tables 1, 2.

Among the ReaChR analogs, the A2-ReaChR formed from 2b, having an extra double bond in the cyclohexene ring compared to retinal, showed $\lambda_{\text{max}}$ at 580nm with a 42 nm red-shift compared to the A1-ReaChR (native ReaChR, $\lambda_{\text{max}}$ at 538 nm). In using A1 analogs having the same conjugated system of A2, the absorption maxima of three new ChRs (A1-6ex-ReaChR [570 nm], A1-10ex-ReaChR [569 nm] and A1-14ex-ReaChR [546 nm]) were shorter than that of A2-ReaChR. When A2 analogs, possessing a π-electron system elongated by one more double bond than A2, were used as chromophore, the different results were obtained according to the chromophore. Thus, the A2-14ex-ReaChR showed $\lambda_{\text{max}}$ at 579 nm, which was almost the same as A2-ReaChR, whereas A2-6ex-ReaChR and A2-10ex-ReaChR exhibited $\lambda_{\text{max}}$ at 607 nm and 596 nm, respectively, which are longer than that of A2-ReaChR and their red-shift values were 69 nm and 58 nm relative to the A1-ReaChR.

In addition, the shape of absorption spectra broadened the band width. As a result, the long-wavelength spectral boundary of A2-6ex-ReaChR and A2-10ex-ReaChR were extended to 672 nm (shift of 84 nm) and 673 nm (shift of 85 nm), respectively. These red-shift values were greater than that of A2-14ex-ReaChR (68 nm). It is noteworthy that the position of inserted one double bond was very important in causing the effective red-shift in the absorption spectra.

Similar tendencies were also observed in ChrimsonR analogs. In the case of A1 analogs having the same conjugated system of A2, absorption maxima of all new ChRs (A1-6ex-ChrimsonR [625 nm] and A1-10ex-ChrimsonR [636 nm], A1-14ex-ChrimsonR [597 nm]) were the same or of shorter wavelengths than that of A2-ChrimsonR (636 nm). When A2 analogs were used as the chromophore, the $\lambda_{\text{max}}$ of A2-14ex-ChrimsonR (634 nm) was shorter than that of A2-ChrimsonR; on the other side, the $\lambda_{\text{max}}$ of A2-6ex-ChrimsonR (651 nm) and A2-10ex-ChrimsonR (654 nm) were slightly red-shifted (15 and 18 nm) compared to that of A2-ChrimsonR. The red-shift values relative to A1-ChrimsonR (native ChrimsonR, 587 nm) were moved 64 and 67 nm, respectively. Furthermore, the width expansion of the absorption spectra in A2-6ex-ChrimsonR and A2-10ex-ChrimsonR were more remarkable than for A2-14ex-ChrimsonR, with the long-wavelength spectral boundaries being extended to 746 nm (shift of 102 nm) and 756 nm (shift of 112 nm), respectively. In both cases, these positions were over 700 nm and reached into the far-red region (710–850 nm).

The capability of a new ChR to absorb the visible light was estimated by the value of relative absorbance ($\text{Abs}_{\text{max}}/\text{Abs}_{278\text{nm}}$). While the relative absorbance depends on the binding efficiency of the analogs, it would be affected by the difference in the extinction coefficients and/or stability of formed pigments.\textsuperscript{18} All analogs seemed to bind with opsin smoothly to afford new ReaChRs (Table 1). In contrast, in the case of ChrimsonR, a remarkable difference was observed. Thus, A2 and its analogs gave values similar to retinal, whereas the value of A1 analogs were significantly decreased (Table 2). These results suggested that the binding pocket in ChrimsonR was not suitable for the A1 analogs due to their steric hindrance. Such steric hindrances might be reduced in the case of A2 analogs by the introduction of a double bond into the cyclohexene ring. To clarify these observations, the interactions between chromophore and amino acid residues of opsin are now in progress using computational docking simulations.

**Conclusion**

We designed and prepared six new kinds of chromophores, with one double bond inserted into the polyene side chain of retinal (A1) or 3,4-didehydroretinal (A2) for producing the
red-shifted ChRs. All chromophores bound with ReaChR and ChrimsonR to afford the new red-shifted ChRs. Furthermore, the absorption spectra of all new ChRs were broadened towards the long-wavelength side and remarkable red-shifts were observed. Among the new ChRs, A2-10ex-ChrimsonR exhibited the longest red-shift effect in its absorption spectrum. Therefore, A2-10ex chromophore seemed to be the most promising for further development of a new red-shifted ChR. Photochemical behaviors of these new ChRs are in progress and will be published elsewhere.

**Table 1. Absorption Characteristics of ReaChR Analogs**

|                  | A1-Analog | A2-Analog |
|------------------|-----------|-----------|
|                  | A1       | 6ex       | 10ex      | 14ex      |
| λ<sub>max</sub> (nm) | 538      | 570       | 569       | 546       |
| Shift (Δ nm)<sup>a</sup> | —        | 32        | 31        | 8         |
| 1/e λ<sub>max</sub> (nm) | 588      | 632       | 635       | 620       |
| Shift (Δ nm)<sup>a</sup> | —        | 44        | 47        | 32        |
| Relative absorbance<sup>b</sup> | 0.35  | 0.20       | 0.21       | 0.20       |

<sup>a</sup> Shift from A1-ReaChR. <sup>b</sup> The values of Abs<sub>max</sub>/Abs<sub>278 nm</sub>.

**Table 2. Absorption Characteristics of ChrimsonR Analogs**

|                  | A1-Analog | A2-Analog |
|------------------|-----------|-----------|
|                  | A1       | 6ex       | 10ex      | 14ex      |
| λ<sub>max</sub> (nm) | 587      | 625       | 636       | 597       |
| Shift (Δ nm)<sup>a</sup> | —        | 38        | 49        | 10        |
| 1/e λ<sub>max</sub> (nm) | 644      | 700       | 711       | 660       |
| Shift (Δ nm)<sup>a</sup> | —        | 56        | 67        | 16        |
| Relative absorbance<sup>b</sup> | 0.26  | 0.08       | 0.20       | 0.12       |

<sup>a</sup> Shift from A1-ChrimsonR. <sup>b</sup> The values of Abs<sub>max</sub>/Abs<sub>278 nm</sub>.

**Fig. 2. UV-Vis Absorption Spectra of ReaChRs**

Bearing (A) A1 or its analogs and (B) A2 or its analogs. With A1 or A2 (black, 1), 6ex (red, 2), 10ex (green, 3) and 14ex (blue, 4). The contribution of ReaChR regenerated by the endogenous A1 in HEK293 cells was subtracted. All the absorption spectra were normalized to 1.0 at 278 nm.

**Fig. 3. UV-Vis Absorption Spectra of ChrimsonRs**

Bearing (A) A1 or its analogs and (B) A2 or its analogs. With A1 or A2 (black, 1), 6ex (red, 2), 10ex (green, 3) and 14ex (blue, 4). The contribution of ChrimsonR regenerated by the endogenous A1 in HEK293 cells was subtracted. All the absorption spectra were normalized to 1.0 at 278 nm.

Experimental

UV-vis spectra were recorded on a JASCO Ubest-55 or JASCO V-650 instrument. IR spectra were measured on a
PerkinElmer, Inc. FT-IR spectrometer, model Paragon 1000 or Horiba FT-IR spectrometer, model FT-720. NMR spectra were obtained on a Varian Mercury-300, a VXR-500 or a Bruker Avance III 600 superconducting FT-NMR spectrometer. Chemical shifts (δ) are reported in ppm relative to tetramethylsilane as an internal reference (CDCl3; δ = 0 ppm for 1H) or residual solvent signals (CDCl3; δ = 77 ppm for 13C). Mass spectra were taken on a Thermo Fisher Scientific Exactive spectrometer. Column chromatography was performed using Kanto Silica Gel 60N (spherical, neutral). All reagents were used as obtained commercially unless otherwise noted.

**General Procedure for the Synthesis of C2-Elongated Conjugate Ester by Horner–Wadsworth–Emmons Reaction (GP1)** To a stirred solution of triethyl phosphonoacetate (C2-phosphonate, 2.5–5.0 eq.) in dry tetrahydrofuran (THF) (0.2 M/phosphonate) was added a solution of NaN(TMS)2 (2.5–5.0 eq.) in dry tetrahydrofuran (THF) (0.2 M/ester) was added DIBAL (3.5 eq., 1.0 M in hexane) in vacuo. The resultant allyl alcohol was used for the next MnO2 oxidation step without further purification.

To a solution of allyl alcohol in Et2O (0.04 M) was added MnO2 (10 times weight relative to ester) at r.t. and was stirred for 14–18 h. The resulting mixture was filtered through Celite with Et2O, and evaporated in vacuo. The residual allyl alcohol was used for the next MnO2 oxidation step without further purification.

(E)-3-(2,6,6-Trimethylcyclohexen-1-yl)-2-propan-2-ol (7a) According to GP1, ethyl (E)-3-(2,6,6-trimethylcyclohexen-1-yl)-2-propan-2-ol (1.37 g, 94%) was obtained from β-cyclotriol (6a: 1.00 g, 6.57 mmol), triethyl phosphonoacetate (7.36 g, 32.9 mmol) and n-BuLi (20.5 mL, 1.57M in hexane, 32.2 mmol). Eluent: hexane/EtOAc = 9: 1. According to GP3, 7a (428 mg, 37%, in 3 steps from 6a) was obtained from ethyl (E)-3-(2,6,6-trimethylcyclohexen-1-yl)-2-propanol (1.37 g, 6.18 mmol), LiAlH4 (469 mg, 12.4 mmol) and MnO2 (85% purity, 14.0 g, 137 mmol). Eluent: hexane/EtO = 4: 1. Spectral data were identical with those of the literature.20

7a: Pale yellow oil. IR (CHCl3) v<sub>max</sub> cm<sup>−1</sup>: 2928, 1676, 1605. 1H-NMR (300 MHz, CDCl3) δ: 9.51 (1H, d, J = 7.8 Hz), 7.26 (1H, d, J = 16.2 Hz), 6.17 (1H, dd, J = 16.2, 7.8 Hz), 2.10 (2H, t, J = 6.0 Hz), 1.79 (3H, s), 1.70–1.53 (2H, m), 1.53–1.47 (2H, m), 1.09 (6H, s). 13C-NMR (75 MHz, CDCl3) δ: 194.95, 152.99, 139.80, 136.25, 132.79, 40.17, 34.26, 34.24, 28.90, 22.02, 18.94. (E)-3-(2,6,6-Trimethylcyclohexa-1,3-dien-1-yl)-2-propanone (7b) According to GP1, ethyl (E)-3-(2,6,6-trimethylcyclohexa-1,3-dien-1-yl)-2-propanone (11.9 g, 85% purity, 119 g, 1.16 mol). Eluent: hexane/Et2O = 20: 1. According to GP3, 7b (3.74 g, 35%, in 3 steps from 6b) was obtained from ethyl (E)-3-(2,6,6-Trimethylcyclohexa-1,3-dien-1-yl)-2-propanone (11.9 g, 53.8 mmol), LiAlH4 (5.11 g, 135 mmol) and MnO2 (E)-3-(2,6,6-Trimethylcyclohexa-1,3-dien-1-yl)-2-propanone (11.9 g, 1.16 mol). Eluent: hexane/EtO = 10: 1

7b: Yellow solid. IR (CHCl3) v<sub>max</sub> cm<sup>−1</sup>: 2962, 1670, 1607. 1H-NMR (300 MHz, CDCl3) δ: 9.54 (1H, d, J = 7.8 Hz), 7.26 (1H, d, J = 16.2 Hz), 6.26 (1H, dd, J = 16.2, 7.8 Hz), 5.99–5.89 (2H, m), 2.15 (2H, d, J = 3.3 Hz), 1.94 (3H, s), 1.11 (6H, s). 13C-NMR (75 MHz, CDCl3) δ: 194.35, 151.23, 135.69, 135.09, 131.62, 129.68, 129.64, 140.17, 34.01, 25.63, 20.41. High resolution electrospray ionization (HR-ESI) MS Calcd for C12H13O [M + H]<sup>+</sup>: 177.1274. Found: 177.1273.

(2E,4E,6E)-3-Methyl-7-(2,6,6-trimethylcyclohexen-1-yl)-2,4,6-heptatrien-1-al (8a) According to GP2, ethyl (E)-3-methyl-7-(2,6,6-trimethylcyclohexen-1-yl)-2,4,6-heptatrienol (628 mg, 90%) was obtained from 7a (428 mg, 2.40 mmol), triethyl 3-methyl-4-phosphonoacetate (1.77 g, 7.20 mmol), n-BuLi (4.50 mL, 1.57M in hexane, 7.07 mmol) and DMPU (1.10 mL, 9.59 mmol). Eluent: hexane/ EtOAc = 9: 1. According to GP3, 8a (360 mg, 58%, in 3 steps from 7a) was obtained from ethyl (2E,4E,6E)-3-methyl-7-(2,6,6-trimethylcyclohexen-1-yl)-2,4,6-heptatrienoate (628 mg, 2.18 mmol), LiAlH4 (165 mg, 4.35 mmol) and MnO2 (85% purity, 630.15 mmol). Eluent: hexane/EtO = 4: 1. This compound is known and main signals of 1H-NMR were reported in the literature.21

8a: Pale yellow oil. IR (CHCl3) v<sub>max</sub> cm<sup>−1</sup>: 2932, 1656, 1594. 1H-NMR (300 MHz, CDCl3) δ: 10.10 (1H, d, J = 8.1 Hz), 6.82 (1H, dd, J = 15.6, 10.5 Hz), 6.46 (1H, d, J = 15.6 Hz), 6.31 (1H, J/2 = 15.6 Hz).
According to GP2, ethyl (2E,4E,6E)-3-methyl-7-(2,6,6-trimethylcyclohexa-1,3-dien-1-yl)-2,4,6-heptatrien-1-ol (8b) was obtained from ethyl (2E,4E,6E)-3-methyl-7-(2,6,6-trimethylcyclohexa-1,3-dien-1-yl)-2,4,6-heptatrienioate (7.98 g, 95%) was obtained from 7b (5.20 g, 29.5 mmol), triethyl 3-methyl-4-phosphono- 


crotonate (11.6 g, 44.2 mmol), n-ButLi (280.0 mL, 1.58 M in hexane, 44.2 mmol) and DMPU (7.11 mL, 59.0 mmol). Eluent: hexane/EtOAc = 30:1.

According to GP3, 8b (4.72 g, 66%), in 3 steps from 7b was obtained from ethyl (2E,4E,6E)-3-methyl-7-(2,6,6-trimethylcyclohexa-1,3-dien-1-yl)-2,4,6-heptatrienioate (7.98 g, 27.9 mmol), DIBAL (100 mL, 1.0 M in hexane, 100 mmol) and MnO₂ (85% purity, 79.8 g, 780 mmol). Eluent: hexane/EtOAc = 4:1.

8b: Yellow solid. IR (CHCl₃) ʋmax cm⁻¹: 3013, 1651, 1592.

1H-NMR (300 MHz, CDCl₃) δ: 10.11 (1H, d, J = 8.1 Hz), 6.83 (1H, dd, J = 15.6, 9.9 Hz), 6.48 (1H, d, J = 15.6 Hz), 6.36 (1H, dd, J = 15.6, 9.9 Hz), 6.35 (1H, d, J = 15.6 Hz), 5.96 (1H, d, J = 8.1 Hz), 5.87 (1H, d, J = 9.6 Hz), 5.79 (1H, dt, J = 9.6, 4.2 Hz), 2.31 (3H, s), 2.10 (2H, dd, J = 4.2, 0.9 Hz), 1.90 (3H, s), 1.06 (6H, s). 13C-NMR (75 MHz, CDCl₃) δ: 191.00, 154.66, 137.48, 134.82, 135.43, 131.81, 129.84, 129.51, 129.66, 39.22, 33.86, 26.71, 20.39, 12.92. HR-ESIMS Calcd for C₂₂H₃₁O [M⁺]: 243.1743. Found: 243.1745. 20a: Pale yellow oil. IR (CHCl₃) ʋmax cm⁻¹: 2923, 1663, 1590.

1H-NMR (300 MHz, CDCl₃) δ: 9.61 (1H, d, J = 8.1 Hz), 7.53 (1H, dd, J = 15.0, 12.0 Hz), 6.50 (1H, d, J = 15.9 Hz), 6.29 (1H, d, J = 12.0 Hz), 6.20 (1H, d, J = 15.9 Hz), 6.15 (1H, dd, J = 15.0, 8.1 Hz), 2.09 (3H, s), 2.04 (2H, t, J = 6.0 Hz), 1.72 (3H, s), 1.69–1.58 (2H, m), 1.51–1.42 (2H, m), 1.04 (6H, s).

13C-NMR (75 MHz, CDCl₃) δ: 193.95, 148.29, 147.01, 137.60, 136.64, 131.73, 131.82, 130.95, 128.65, 39.80, 34.48, 33.43, 29.18, 22.01, 19.34, 13.37.
According to GP3, 4a (235 mg, 36%, in 3 steps from 10a) was obtained from ethyl (2E,4E,6E,8E,10E)-3,9-dimethyl-11-(2,6,6-trimethyl-cyclohexen-1-yl)-2,4,6,8,10-undecapentaenoate (709 mg, 2.00 mmol), LiAlH₄ (152 mg, 4.01 mmol) and MnO₂ (85% purity, 7.10 g, 69.4 mmol). Eluent: hexane/Et₂O = 4:1.

4a: Orange solid. UV-vis (EtOH) λ_max 403 nm (ε = 48000).

IR (CHCl₃) ν_max cm⁻¹: 2926, 1667, 1603, 1559. ¹H-NMR (600 MHz, CDCl₃) δ: 9.61 (1H, d, J = 7.8 Hz), 7.51 (1H, dd, J = 15.0, 12.0 Hz), 6.92 (1H, dd, J = 14.4, 11.4 Hz), 6.38 (1H, d, J = 15.0 Hz), 6.35 (1H, d, J = 12.0 Hz), 6.29 (1H, d, J = 16.2 Hz), 6.18 (1H, dd, J = 12.6, 7.8 Hz), 6.17 (1H, d, J = 12.6 Hz), 6.15 (1H, d, J = 16.2 Hz), 2.11 (3H, s), 2.04 (2H, t, J = 6.0 Hz), 2.01 (3H, s), 1.72 (3H, s), 1.65–1.58 (2H, m), 1.49–1.45 (2H, m), 1.04 (6H, s). ¹³C-NMR (150 MHz, CDCl₃) δ: 193.56, 147.64, 146.68, 139.45, 137.75, 137.32, 135.51, 130.77, 130.15, 129.96, 129.89, 128.72, 128.48, 39.64, 34.29, 33.15, 28.98, 21.77, 19.22, 13.29, 12.93.

(2E,4E,6E,8E,10E)-5,9-Dimethyl-11-(2,6,6-trimethyl-cyclohexa-1,3-dien-1-yl)-2,4,6,8,10-undecapentaen-1-al (A2-14ex, 5b) According to GP1, ethyl (2E,4E,6E,8E,10E)-5,9-dimethyl-11-(2,6,6-trimethyl-cyclohexa-1,3-dien-1-yl)-2,4,6,8,10-undecapentaenoate (115 mg, 85%) was obtained from 3,4-didehydroretinal (2b) (109 mg, 0.387 mmol), triethyl phosphonoacetate (130 mg, 0.581 mmol) and NaH (60%, 23.2 mg, 0.581 mmol). Eluent: hexane/Et₂O = 20:1.

According to GP3, 5b (64.1 mg, 54%, in 3 steps from 2b) was obtained from ethyl (2E,4E,6E,8E,10E)-5,9-dimethyl-11-(2,6,6-trimethyl-cyclohexa-1,3-dien-1-yl)-2,4,6,8,10-undecapentaenoate (115 mg, 0.327 mmol), DIBAL (1.31 mL, 1.0 M in hexane, 1.31 mmol) and MnO₂ (85% purity, 11.5 g, 11.2 mmol). Eluent: hexane/Et₂O = 5:1. Spectral data were shown in the literature.

Preparation of ChR Analogs and Spectroscopy In this study, two ChR variants, ReaCh (NCBI accession number, KF448069)¹¹ and ChrimsonR (Chrimson (KF992960) K176R mutant)⁹ were used. These ChR variants were expressed by HEK293 system as reported previously.¹⁰ Briefly, the cDNA of these ChR variants were inserted into the mammalian expression vector pCAGGS with the epitope sequence of anti-bovine rhodopsin monoclonal antibody rhod14 (ETSQVAPA) on the C-terminus. HEK293 cells transfected with the plasmid bearing the ChR gene were collected after incubation for 2 d, and suspended in Buffer P (50 mM N-(2-hydroxyethyl)piperazin-N’-2-ethanesulfonic acid (HEPES) and 140 mM NaCl, pH 6.5). Retinal or analog dissolved in ethanol was added to the sample at final concentration of 20 µM. After overnight incubation at 4°C, the regenerated pigments were extracted by 1% n-dodecyl-β-d-maltoside (DDM) in Buffer P, and purified by rhod14-antibody conjugated agarose column. The final solvent of the sample was Buffer P with 0.02% DDM at pH 6.5. All the procedures above were performed under deep-red light.

UV-visible absorption spectra of pigments were measured by UV-visible spectrophotometer (Shimadzu UV-2400), which is equipped with a water circulation system to the cell holder to keep the sample at 4°C. HEK293 cells produce A1,⁹ which generates approx. 10% (ReaChR) or approx. 50% (ChrimsonR) of fully regenerated pigment with excess A1. To show the pure spectra of new ChRs, the contribution of ChR regenerated by the endogenous A1 was subtracted from the sample spectra, and the absorption spectra were normalized to absorbance 1.0 at 278 nm.

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Conflict of Interest The authors declare no conflict of interest.

Supplementary Materials The online version of this article contains supplementary materials.

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