NEW FIND OF ZUNYITE IN ADVANCED ARGILLIC ALTERATION OF RHYOLITES, KOS ISLAND, SOUTH AEGEAN VOLCANIC ARC, GREECE

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ABSTRACT

Zunyite \([\text{Al}_{13}\text{Si}_{5}\text{O}_{20}(\text{OH},\text{F})_{18}\text{Cl}]\), an extremely rare mineral, was found as an accessory mineral in highly altered rhyolitic rocks at Kefalos, southwestern Kos, and in a completely kaolinized rhyolitic dyke at Asfendiou, northeastern Kos. It is associated with dickite, kaolinite, quartz and minor pyrophyllite. Kos Island contains both Tertiary and Quaternary volcanic rocks.

Zunyite has been identified by XRD and studied by SEM, EDS, Raman spectroscopy and FTIR spectroscopy. The presence of zunyite, in the highly altered samples is significant for the determination of the conditions of the hydrothermal alteration that took place.

Zunyite crystals, are up to 90 µm in diameter but usually are less than 10 µm; the crystals are partly covered by kaolinite layers. EDX analyses on the surface of zunyite revealed \(\text{Al}_2\text{O}_3/\text{SiO}_2\) values raging from 2.6 to 2.7.

The occurrence of zunyite suggests incorporation of hydrothermal fluids rich in \(\text{F}^–\) and \(\text{Cl}^–\), thus indicating contamination by seawater.

1 INTRODUCTION

Zunyite \([\text{Al}_{13}\text{Si}_{5}\text{O}_{20}(\text{OH},\text{F})_{18}\text{Cl}]\) is an extremely rare mineral, which is generally found either as an alteration product of feldspars in silicified igneous rocks, as a gangue in metalliferous veins or as an alteration product in advanced argillic alteration.

The mineral zunyite \([\text{Al}_{13}\text{Si}_{5}\text{O}_{20}(\text{OH},\text{F})_{18}\text{Cl}]\) was originally described together with guitermanite, a sulfide of lead and arsenic nowadays known as jordanite, from the Zuni mine on Anvil Mountain, Silverton, San Juan Country, Colorado and named for the Zuni mine (Kloprogge & Frost 1999). Zunyite, is an extremely rare mineral, characteristic of certain hypogene hydrothermal assemblages, forming in a narrow range of physical conditions of temperature and \(\text{pH}\) (Inoue 1995). It possesses not only an unusual combination of an \(\text{Al}_{13}\text{O}_{24}(\text{OH})_{24}\) Keggin structure and a \(\text{Si}_5\text{O}_{16}\) pentamer unit first established by Pauling (1933), but it is also one of the few minerals containing as many as three different volatile components, viz \(\text{H}_2\text{O}, \text{F}\) and \(\text{Cl}\) (Louisnathan & Gibbs 1972). The \(\alpha\) Keggin unit of zunyite is fluorine-substituted in the intertrimeric \(O(4)\) site, leading to the creation of crystallographically unequivocal sites for fluorine and lowering the symmetry of the central tetrahedral site (Zhou & Sherriff 2003). The issue of \(\text{Al}:\text{Si}\) order has been a topic of controversy for many years. Zagals’ skaya & Belov (1964) suggested that \(\text{Si}\) should be in the center of the Keggin structure. Refinements of Louisnathan & Gibbs (1972) and Baur & Ohta (1982) disproved this arrangement, but showed that excess \(\text{Al}\) could be accommodated in the center of the pentamer.

The presence, in trace amounts, of alunite, in highly altered rhyolitic rocks at Kefalos, a mineral much more common than zunyite in hydrothermally altered formations, indicates that the hydrothermal fluids were poor in \(\text{S}^2\).-2

The mineral zunyite is so rare in nature that any information regarding a new occurrence seems worthy of record.
2 GEOLOGICAL SETTING AND MATERIALS

The mineral zunity was found as an accessory mineral in highly altered rhyolitic rocks at Kefalos, southwestern Kos, and in a completely kaolinized rhyolitic dyke at Asfendiou, northeastern Kos (Fig. 1). It is associated with dickite, kaolinite, quartz and minor pyrophyllite. Kos Island contains both Tertiary and Quaternary volcanic rocks. The Quaternary volcanism produced a successive series of rhyolitic-dacitic volcanic rocks and pyroclastic flows (Boven et al. 1987, Dalambakis 1987). The kaolin occurrences of the Kefalos peninsula, SW Kos Island, are associated with rhyolitic rocks of Pliocene age (Boven et al. 1987). These rocks exhibit perlitic texture and are composed of quartz, Na-plagioclase, K-feldspar and volcanic glass. The kaolin occurrences are generally white in colour, but frequently are also stained reddish by iron oxides; they extend over an area of about 0.5 km² (Papoulis & Tsolis-Katagas 2001, Papoulis 2003). Kaolin occurrences of Asfendiou area are products of hydrothermal alteration of a rhyolitic dyke. The parent rock is almost completely altered but rare remnants occur.

Figure 1. a). Geographic position of Kos Island, in the Southeastern Aegean Sea. b). Geological sketch map of Kos Island (after Triantaphyllis 1994, 1998).
3 METHODS

Mineralogical compositions of bulk rock and of clay fractions (<2 µm) of the samples were determined by X-ray diffraction, using a Philips diffractometer PW1050/25, with Ni-filtered CuKα radiation. Powders from oriented clay-aggregate samples were scanned at $1^\circ/\text{min}$ from 3 to 60°. Clay minerals were identified from three XRD patterns (air-dried at 25°C, ethylene-glycolated, and heated at 490°C for 2h) of the clay-size fraction (<2 µm) extracted by the standard sedimentation technique in deionized water.

Selected samples were examined under shortwave and longwave ultraviolet light.

Mineral textures, morphology and chemical composition of alteration products were determined at the Laboratory of Electron Microscopy and Microanalysis, University of Patras using a Scanning Electron Microscope (SEM, JEOL 6300) equipped with EDS and WDS spectrometers and a THETA software. The chemical composition of zunyite was determined using EDX analyses.

The FT-Raman spectra were obtained with a power of 3 mW on the specimen for a total integration time of 12 s; a viewing screen connected to the microscope offered good sample positioning, good laser beam focusing, and direct surface inspection. The scattered light was filtered (HNF-514-1.0 from Kaiser Optical Systems), to remove the elastic Rayleigh scattering. The T-64000 (Jobin Yvon) Raman system, equipped with a Spectraview-2D liquid N2-cooled CCD detector was used, in the single spectograph configuration, to disperse and detect the Raman signal. The spectral resolution for the Raman spectra used for molecular orientation was ~5 cm$^{-1}$. The spectral window was centered at 3600 cm$^{-1}$ (in the Stokes region). The Raman spectrum has been recorded with the FT-Raman FRA-106-S module of an Equinox 55 FTIR spectrometer of Bruker.

Infrared spectra were measured in the specular reflectance mode on a Bruker vacuum spectrometer (IFS 113v) equipped with a near-normal incidence (11°) reflectance accessory. A KRS-5 wire grid polarizer was positioned before the sample to polarize the infrared radiation perpendicular to the plane of incidence. The polarizer remained fixed and the sample was rotated by 90° to obtain reflectance spectra with different polarization directions. The reflectance spectrum of a high reflectivity aluminum mirror was measured with the polarizer in place and the same instrument settings and used as reference spectrum. All spectra were measured at room temperature and represent the average of 200 scans at 2 cm$^{-1}$ resolution.

4 RESULTS

Zunyite in the studied areas was recognized from its XRD pattern in highly altered rhyolitic rocks (Figs. 2a, b). Dickite, kaolinite, zunyite and quartz coexist in these highly altered samples. Most of the samples containing zunyite examined under shortwave ultraviolet light fluoresce red. The same samples under longwave ultraviolet light fluoresce in various colours (purple, blue flecks). SEM observations show that zunyite crystals are up to 90 µm in diameter (Fig. 3), but usually are less than 10 µm. Zunyite identification is based on its cubic crystal habit and EDX analyses (Fig. 4). EDX analyses and spectra taken from the surface of zunyite revealed Al$_2$O$_3$/SiO$_2$ values ranging from 2.6 to 2.7. Its crystals are usually partly covered by kaolinite layers (Figs. 3, 4c, d) and EDX analyses on covered surfaces of zunyite show lower Al$_2$O$_3$/SiO$_2$ ratios (< 2.5) (Figs. 4c, d), due to the presence of kaolinite.

Representative FTIR spectra of highly altered samples from Kefalos, in the region 400-4200 cm$^{-1}$ are shown in Fig 5a. In the highly altered samples the position and the relative intensity of the bands, especially in the OH stretching region, indicate the presence of dickite and minor kaolinite (Fig. 5a). The bands of dickite in the OH stretching region are not well formed showing poorly ordered dickite. This is also attributed to the presence of zunyite bands (3649 cm$^{-1}$, Turco 1962, 3625 cm$^{-1}$, Gadsden 1975) in the same area with dickite bands. The other characteristic bands of zunyite 725, ~900, 1000, ~1176 cm$^{-1}$ (Turco 1962, Gadsden 1975, Kloprogge & Frost 1999) overlap with
the bands of dickite and kaolinite. The FTIR spectra of the highly altered samples of Asfendiou (Fig. 5b) are similar to the highly altered samples from Kefalos.

The FT-Raman spectra of the low frequency region of representative completely kaolinized rocks from Kos, shown in Figs. 6a, b confirmed also the presence of zunyite. The band positions (cm$^{-1}$) of zunyite ~200, 355, 395, 520 and 615 are very close to the positions of zunyite suggested by Kloprogge & Frost (1999).

In the highly altered samples from Kefalos (Fig. 6a) dickite coexists with zunyite; minor kaolinite is also present. The FT-Raman spectra, in the region 100-1200 cm$^{-1}$, of highly altered samples from Asfendiou show also the coexistence of zunyite, kaolinite and dickite (Fig. 6b).

Figure 2. X-Ray powder diffraction patterns of highly altered rhyolitic rocks from Kefalos (a) and Asfendiou (b). Z=Zunyite, D=dickite, K=kaolinite, Q=quartz.

Figure 3. SEM microphotographs of a hydrothermal kaolin sample from Kos. (a) The single crystal in the center is zunyite, (b) Magnification of the box.
Figure 4. Energy dispersive X-Ray spectra (EDX) of zunyite (a, b) and zunyite crystals partly covered by kaolinite layers (c, d).

Figure 5. FTIR spectra of highly altered samples K35 (a) and K48 (b). K = kaolinite, Q = quartz, D = dickite, Z = zunyite.
DISCUSSION

The coexistence of zunyite, kaolinite, dickite and traces of pyrophyllite, is not feasible in a wide range of conditions. According to Hsu (1985) both F and Cl are indispensable to the formation of zunyite and yet higher F activities destabilize it. This assemblage reflects a late hydrothermal event, which resulted in advanced argilic alteration of rhyolites and a subsequent kaolin formation. According to Inoue (1995), Dill et al. (1997) and Willan & Armstrong (2001) the assemblage Dickite + Zunyite + Quartz + Kaolinite ± Pyrophyllite is stable at a temperature range between 250°C and 290°C, at low pH (<3.5).

The occurrence of zunyite, in the kaolins of Kos island, implies incorporation of hydrothermal fluids rich in F− and Cl−, indicating also contamination by seawater. Zunyite typically coexists with dickite in some advanced argilic alteration zones (Dill et al. 1997, Sillitoe et al. 1998). Alunite is a common mineral in advanced argilic alteration zones and forms at similar conditions with zunyite. However, the formation of alunite requires hydrothermal fluids rich in S2−, hence the presence of alunite in trace amounts in Kos assemblages indicates the involvement of S2−-poor hydrothermal fluids. Papoulis & Tsolis-Katagas, (2001) argue that the advanced argilic alteration zone at Kefalos extends over small distances around faults. The alteration is presumably associated with these faults as the hydrothermal fluids were channeled along them.

CONCLUSIONS

Zunyite was found as an accessory mineral in highly altered rhyolitic rocks at Kefalos, southwestern Kos, and in a completely kaolinized rhyolitic dyke at Asfendiou, northeastern Kos. It is associated with dickite, kaolinite, quartz and pyrophyllite. Zunyite crystals, are up to 90 µm in diameter but usually are less than 10 µm; the crystals are partly covered by kaolinite layers and their identification under SEM observation is based on their cubic crystal habit and EDX analyses. EDX analyses on the surface of zunyite revealed Al2O3/SiO2 values ranging from 2.6 to 2.7. FT-Raman spectra and XRD studies confirm also the presence of zunyite. The occurrence of zunyite suggests incorporation of hydrothermal fluids rich in F− and Cl−, thus indicating contamination by seawater. The presence, in trace amounts, of alunite, a mineral much more common than zunyite in hydrothermally altered formations, indicates that the hydrothermal fluids were poor in S2−.
The frequent occurrence of zunyite in the altered felsic volcanic rocks of Kos reported here is thus indicative of unusual solution behavior during alteration.

7 ACKNOWLEDGEMENTS

The authors wish to thank Dr. G. Voyiagis of the Laboratory of molecular spectroscopy, Research Center of High Temperature and Chemical Processes, University of Patras, for his help with FTIR, FT-Raman and Raman spectroscopy and Mr. V. Kotsopoulos, of the Laboratory of Electron Microscopy and Microanalysis, University of Patras, for his help with the SEM photomicrographs. The first author is thankful to the State Scholarship Foundation of Greece for the financial support during his Ph.D. study.

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