Antimicrobial Activity of AgNPs Synthesized Via Green Approach by Using Flowers of Bistorta macrophylla herb of Tunghanth Himalaya Region

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Abstract: Green synthesis of nano structured materials is an emerging field for the researchers working in the field of chemical, physical, life sciences, engineering and medical sciences, for the welfare of society. In the present study green synthesis of silver nanoparticles (AgNPs) was performed by using flowers of Bistorta macrophylla, a medicinal herb of polygonaceae family of Garhwal Himalayas. For green synthesis of AgNPs, the aqueous flowers extract and the aqueous solution of AgNO₃ (of 5 milli molar concentration) were well mixed in ratios of 1:9. The metal ion present in the solution reduced Ag⁺¹ to Ag⁰ within three to four days, which was characterized by color change observation, UV, SEM-EDX, XRD, TEM etc. The absorption maxima of AgNPs of Bistorta macrophylla flowers (BMF) was obtained in the range of 451–454 nm for 5 mM concentration of AgNO₃, and the SEM images revealed that the average clustersize of synthesized Ag nanoparticles from 5BMF₁:9 ranges from 20 μm–23 μm, while the EDX analysis four elements show their presence as Ag (51.44 weight %), C (22.45 weight %), Cl (1.18 weight %) and O (22.24 weight%) in the synthesized AgNPs of 5BMF₁:9. XRD revealed that the crystals nature of synthesized AgNPs were cubic, and TEM results shows that the biosynthesized nanoparticles are agglomerated and irregular shaped in which size ranging from 12 nm to 38 nm. Synthesized AgNPs were also found to be active for potent antibacterial activity against various pathogenic and harmful bacterial strains of humans like Pseudomonas Sp. (19.5 mm), Bacillus cereus (21 mm) and CoNS (Coagulase Negative Staphylococci) (16.4 mm). Synthesized nanomaterial may be active for efficient clinical applications.

Keywords: Green synthesis • Bistorta macrophylla • Polygonaceae • Antimicrobial • SEM-EDX

Introduction

Nanoparticles can be defined as sub nano sized colloidal structures with particles size between 1 to 100 nm which is rapidly emerging stream nanotechnology (Vyas and Khar, 2002; Sati et al., 2020a) and shows unique properties and application of nanomaterials. The metal nanoparticles like silver, zinc oxide, gold etc. have several applications viz. optics, coating, nanomedicines, super capacitor bio-labeling, batteries, and catalytic agent in reactions, antibacterial materials, chemically stable materials and good electrical conductors (Sharma et al., 2009; Zargar et al., 2014). Green nanotechnology research pathways is an interesting platform for the preparation of nano structured particles as well as nanocomposites (Baskaralingam et al., 2011). Green root synthesis is an eco-friendly, nontoxic, and safe technology for the formation of nano sized materials in the field of material sciences which includes the formation of
nanoparticles by metal reduction (silver) process in the presence of various phyto-constituents which are responsible for the reduction. Effective synthesis of metal nanoparticles (MNPs) is the requirement of present scenario, because synthesized MNPs can be active material for many biological activities (Bunghez et.al., 2011; Thirumurugan, 2010).

The plant species Bistorta macrophylla (Polygonum macrophyllum) belongs to family Polygonaceae, is native to mountain regions of West Bengal, Tibet, Bhutan, South China, Pakistan and North India (Kashmir, Himachal, Uttarakhand, Valley of flower, Tungnath Himalaya). Bistorta macrophylla is a perennial, small, glabrous herb of Tungnath Himalaya. It is also known as kukhri, kutki in locality. The plant has a height upto 10-25 cm long with thick rootstock and long radical leaves at an altitude of 3000-4200 meters (Gaur, 1999), which can be identified by color of flower i.e. dark red purple color.

| Plant Name : | Family : |
|---------------|---------|
| Bistorta macrophylla | Polygonaceae |
| Species : B. macrophylla |
| Local name : Kukhri |
| Habitat : Alpine herbs |

The plant is native from harsh climatic conditions so it can tolerate such harsh temperature. In the phyto-chemical screening of acetone, methanolic, and aqueous extract of B. macrophylla, the presence of glycosides, epicatechin-5-O-beta-D-glucopyranoside, D-glucopyranoside, chlorogenic acid and gallic acid, tannin (67.96g/Kg), saponin (Zhang et.al., 2008) and many compounds have been reported. Phenol contents are highest constituent (98.12g/Kg) in flowers of P. macrophyllum. Significant presence of phenol is responsible for protein precipitation in P. macrophyllum (Wang et.al., 2004). Selected plant has several antioxidant properties and potent antimicrobial activity which may be helpful in generating uncommon compound for the development of novel antibacterial agents by nanoparticles synthesis (Kirtikar and Basu, 2006), which may fight against various human pathogens.

**Materials and Methods**

The flowers (Fig.1) of B. macrophylla (Kukhri) was collected from Tungnath Himalaya region of Garhwal, Uttarakhand and identified by Taxonomist, Department of Botany HNB Garhwal University Srinagar Garhwal, where the voucher specimens (GUH 6894) was deposited (Gaur, 1999).

![Fig.1: Flowers of Bistorta macrophylla (BMF)]

Green Synthesis of Silver Nanoparticles

In green approach synthesis, flowers of B. macrophylla had been used for the synthesis of AgNPs. First of all flowers of selected plant were dried in shade for 12-15 days, and then the well dried flowers of B. macrophylla were crushed in mixer grinder. For the preparation of flowers extract, we took 5 gram powder of dried flowers in 100 ml deionised water in a round bottom flask and put it on heating mental at 70-80°C for 30min. Resultant flowers extract was filtered two times for removal of contaminants.

Freshly prepared flowers extract added into AgNO₃ (Sigma Aldrich, Germany) solution of 5 milli molar concentrations in ratio 1:9, (Bartwal et
The resulting metal nanoparticles (MNP) washed two times using deionised water and dry in shade at room temperature (Bartwal et al. 2020, Sati et.al.,2020b). Then synthesized samples of AgNPs were kept for further analysis.

**Characterization techniques**

Prepared AgNPs were characterized by several techniques viz. Visual examination, UV spectroscopy, SEM-EDX, XRD and TEM to define various parameters of nanoparticles like shape, size, morphology, composition, etc. UV data were taken by Double beam spectrophotometer model 3375 (Electronics India), XRD measurements of prepared sample was taken, at room temperature, by an X-ray diffractometer (PANalytical, X'PERT PRO), using Cu Kα1 radiation of wavelength 1.5405980 Å, in a broad range of 2θ (20° to 70°) at a scanning rate of 6°/min. Surface morphology and grain size of the prepared sample was studied using a SEM (CARL ZEISS, MA15/EVO18), and TEM was taken at HV 120kV (Sati et.al.,2020; Bartwal et al. 2020).

**Antimicrobial activity**

To analyse the antimicrobial activity of synthesised AgNPs against the human pathogenic bacteria viz. *Psudomonas, Bacillus cereus*and CoNS (Coagulase Negative Staphylococci), the well diffusion method (Sati et al. 2020) was employed. AgNPs were dissolved in DMSO. Incubation of petri plates was done at 37°C for 24 hours and by measuring the zone of inhibition around the well, antimicrobial activity was determined. The antibacterial effect of synthesised AgNPs was recorded by the measurement of inhibited zone around the well (Sharma et.al., 2009; Kour et.al. 2018; Paudel et.al., 2018).

**Results and Discussion**

**Visual examination:** The preliminary examination of prepared AgNPs was confirmed on the basis of change in color of solutions. The color change examination of mix solution of *B.macrophylla* flowers (BMF) and AgNO₃ was observed with respect to time (Bartwal et al., 2020).

![Fig.2: Color change observation at different time intervals (in hours) (a) t=0; (b) t=18; (c)t=42; (d) t=66.](image)

On addition of *B. macrophylla* flowers extract to AgNO₃ solution, color of entire solution system had been changed from whitish yellow to brown red followed by dirty green color (see Fig.2), which exhibit the formation of AgNPs i.e. Ag⁺ ions have been converted to elemental Ag⁰ in reaction medium having the size of nano structure range. Obtained results were found similar and were in-accordance to results performed by other researchers (Bartwal et al., 2020; Bagyalakshmi and and Haritha, 2017).

**UV Analysis**

UV analysis shows the formation and completion of formation of AgNPs. AgNPs formation takes place due to surface plasmon resonance (SPR) absorption band due to the combined vibrations of electrons of MNP in resonance with light wave. AgNPs are known to exhibit UV peak in the range of 451-454 nm (Bartwal et al.,2020; Bagyalakshmi and and Haritha, 2017).

The absorption peak of synthesized AgNPs were observed at 451nm for 5mM (BMF₁:9), for flowers of *B. macrophylla* (Fig.3).
XRD Analysis

Fig.4 indicated the diffraction pattern of AgNPs which shows peaks corresponding to elemental Ag (XRD Ag score 81%). XRD peak 20 values observed at 38.117°, 44.279° and 64.428°, with crystalline planes of silver (111), (200) and (220), respectively. The obtained results explain that the produced AgNPs are of 84.260 Å crystal size with cubic crystalline phase, for which the lattice parameteris ‘a’ = 4.0862 Å (Sati et.al., 2020a, 2020b, Bartwal et.al., 2020; Bagyalakshmi and Haritha, 2017).

SEM-EDX Analysis

The SEM is convenient for inspection of grain or cluster structure of synthesized nano material AgNPs. It produces 3D appearing Images of micro structural features for observing surface morphology. The prepared AgNPs nanocomposition of 5BMF₁₉ average grain size was 22.16μm (Sati et.al., 2020a, 2020b), some clusters exceeds more than 40 μm with agglomerated and smooth surfaces (see Fig.6).

Fig.5: EDX Graph with bar diagram

In the EDX analysis, result of synthesized AgNPs of 5BMF₁₉ observed that the significant Ag was present with 51.44% of weight percentage with some chlorine (1.18%), oxygen (24.93%) and carbon (22.45%)[2,12] contaminants in the synthesized nanoparticles (see Fig.5).
TEM Analysis
The surface morphology and size of the produced MNPs were also determined through TEM which is a convenient tool, in which the average particle size was calculated by taking the average size of the smallest and largest particles in the image. In the TEM analysis of AgNPs of 5BMF, average grain size have been observed in range of 12 nm to 38 nm (Bartwal et al., 2020; Bagyalakshmi and Haritha, 2017), while some grains size exceeds upto 40 nm (Fig.7).

Antimicrobial activity

Table 5.1: Antimicrobial activity of AgNPs

| S. No. | Organisms    | Zone of inhibition (mm) | Positive Control (Streptomycin-25µg/ml) |
|--------|--------------|-------------------------|----------------------------------------|
| 1.     | *Bacillus cereus* | 21                      | 22.6                                   |
| 2.     | *Psudomonas*   | 19.5                    | 21                                     |
| 3.     | *CoNS*        | 16.4                    | 20                                     |

Conclusion
The present research work focused on the synthesis of MNPs (AgNPs) using *B. macrophylla* flowers (BMF) extract via green approach of metal reduction process. The phytochemical organic compounds present in the flowers extract of *B. macrophylla* act as a strong bio-reducing and stabilizing agent during the synthesis of AgNPs. Furthermore, the AgNPs production is also confirmed by different spectroscopic techniques. UV analysis shows the absorption maxima peak at 451 nm for 5 mM (BMF) for flowers of *B. macrophylla*, which is confirmation of AgNPs formation. XRD confirms that the synthesized nanoparticles exhibit cubic crystalline structure with average crystal size of 84.260 Å. SEM image shows the agglomerated particles of irregular shapes. TEM analysis confirms the average grain sizes of AgNPs were found in the range 12 nm to 38 nm with spherical smooth morphology. These nanoparticles also
showed potent antimicrobial activity against the human pathogenic bacterial strains. Synthesis of silver NPs based of green chemistry principles is cost effective method which may be used as an alternative of chemical synthesis methods. Synthesized nanomaterial may be useful in pharma industries as development of antimicrobial emulsions, creams and in clinical applications.

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Conflict of interest

There is no conflict of interest. This research work is not funded by any funding agencies.

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