HEPATOPROTECTIVE ROLE OF *CISSUS QUADRANGULARIS* ON LEAD ACETATE-INDUCED LIVER INJURY IN FEMALE WISTAR RATS

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ABSTRACT

**Objectives:** Several heavy metals like lead acetate can accumulate in the body due to exposure to the metal for a prolonged period. One of the possible mechanisms involved with lead toxicity is oxidative stress is for which liver is the target organ. The primary aim of this study was to examine the hepatoprotective role of methanolic stem extract from the herb *Cissus quadrangularis* on induced lead acetate liver injury in female Wistar rats.

**Methods:** The course of the study was for 14 days. The animals were separated into 5 groups: two being control and negative and the other 3 groups based on the dosage of the methanolic extract of the plant was given. The dosage of the plant extract given was once daily for all days of course study. During the last 7 days, lead acetate was injected in the animals (25 mg/kg of body weight). The sacrifice was done 14 days later and the blood and liver samples were taken, which is then used for different antioxidant enzymatic assays.

**Results:** Significant reduced (p < 0.05) antioxidant levels and increased lipid peroxidation levels were observed in lead acetate treated group which was ameliorated by the action of extract from *Cissus quadrangularis* fusions. Histopathological study also supported the finding.

**Conclusion:** The results of the different antioxidant enzymatic assays supported the hepatoprotective role of methanolic stem extract of the plant *Cissus quadrangularis* over induced lead acetate injury in wistar female rats.

**Keywords:** Antioxidant, *Cissus quadrangularis*, Hepatoprotective, Lead acetate.

INTRODUCTION

Lead being a highly toxic metal and poison can affect the nervous, gastrointestinal, renal, and hematic systems over gradual accumulation of the metal in body for years. Lead toxicity can even result in fatal conditions [1]. Lead toxicity occurs when lead is ingested or breathed in along with other dust particles. Lead toxicity can result in various mental and physical symptoms according to the individual’s health status and duration of exposure to the metal, while children being more prone to the condition. Lead has been associated with various forms of cardiovascular disease, nephrotoxicity, cancer, etc. Excess lead in body will reduce the intellectual performance and cognitive development in children, while in adults, it will increase incidence of blood pressure and cardiovascular diseases. Lead is known for inducing over production of reactive oxygen species (ROS) and thereby enhancing lipid peroxidation and unsaturated fatty acid contents in membranes [2]. Enhanced ROS production results in oxidative stress, and ROS itself is a byproduct in many tissue level degenerative reactions and later affects the regular metabolism in cellular level by damaging its components. *Cissus quadrangularis* commonly referred as “bone setter” is native, to countries like, India, Thailand, Africa, etc., is being used for medicinal purposes from centuries back. Its applications include healing bone fracture and tissue repair, preventing osteoporosis, regulation of blood sugar levels, weight loss, irregularity in menstrual period, cholesterol lowering, and various other analgesic, antimicrobial, and anti-inflammatory effects [3,4]. Usually, the dried stem powder is used as the healing agent. Phytochemical analysis of *Cissus quadrangularis* revealed the presence of β carotenoids, for which scientific proofs are available for its hepatoprotection activity [5,6]. Hence, this current study was conducted to examine the hepatoprotective role efficiency of *C. quadrangularis* methanolic stem extract over induced lead acetate toxicity in female Wistar rats.

METHODS

**Chemicals**

Lead acetate and all other reagents and chemicals utilized for the experiment were of analytically graded.

**Preparation of methanolic extract from plant material**

The stems of herb *C. quadrangularis* were collected from Vellore in the month of July. The fresh plant stem is then subjected to shade drying at about 45°C for few days. The stems after drying thoroughly were then powdered (100 g) using an electric grinder. The fine powder obtained after multiple processing of grinding and sieving is then mixed with methanol. The mixture is then left overnight at room temperature. The next day, the solution of plant extract along with methanol is filtered out separately using filter paper from the mixture. The solution obtained is then kept for complete drying at 40°C for about 4-5 hrs in a hot air oven. The extract left behind after drying is then kept at 4°C [6]. Whenever necessary for the experiments, it was used by dissolving in distilled water.

**Animals**

30 female Wistar albino rats with weights varying in the range of 150-200 g were used for the current study. Animals were kept in normal room temperature of 24±2°C in hygienic atmosphere. The animals were fed with normal laboratory diet. The Institutional Animal Ethical Committee of VIT University, Vellore, was the approving authority for the experimental protocol [VIT/IEAC/12/July23/28].

**Experimental design**

Animals were categorized into 5 groups containing 6 animals in each group. Group I (normal control) was treated with normal saline for 14 days. Group II (negative) was treated with lead acetate (25 mg/kg of body weight) from days 8 to 14. Group III (low dosage) was fed with methanolic extract of *C. quadrangularis* (50 mg/kg of the body weight) once daily for 14 days and lead acetate (25 mg/kg of body weight) for
the past 7 days. Group IV (medium dosage) was administered with 100 mg/kg of body weight of methanolic extract of *C. quadrangularis* for 14 days and lead acetate (25 mg/kg of body weight) for past 7 days (once daily). Group V (high dosage) was treated with methanolic extract of *C. quadrangularis* (200 mg/kg of body weight) once daily for 14 days along with lead acetate (25 mg/kg of body weight) for the past 7 days. Lead acetate was injected intraperitonially, and methanolic extract was given orally. Body weights of the rats were assessed on the initial and final days. After 24 hrs of last dosage, the animals were sacrificed by method of decapitation, and trunk blood was collected. Liver tissue was excised instantly in sterile conditions and washed with ice-cold normal saline to remove the fat depositions and blood stains. Later, it was blot dried and the weight of the tissue is taken. These tissue samples were utilized for the histopathological and antioxidant studies.

**Biochemical estimation**

The collected trunk blood was left to clot for few minutes followed with centrifugation at 3000 rpm for 10 minutes at 4°C was done for serum separation. The serum obtained was utilized in estimating the marker enzymes in the liver: Aspartate transaminase (AST) and alkaline phosphatase (ALP). It was done using diagnostic kits which were commercially available from Autospan Diagnostics, India. The total protein estimation was done according to the standard protocol of Lowry method [7].

**Preparation of homogenate**

In ice-cold phosphate buffer (0.1 M with pH 7.4) containing 1.17% KCl, 10% liver homogenate was prepared. A part of this prepared homogenate was utilized in estimating the reduced glutathione (GSH) level and lipid peroxidation. The homogenate remaining was again centrifuged at 10,000 rpm for 10 minutes at a temperature of 4°C. The supernatant obtained is then used to carry out the antioxidant enzymatic assays: Superoxide dismutase (SOD) and catalase (CAT).

**Estimation of reduced GSH**

1 ml of homogenated tissue was added to 1 ml of 5% TCA. Then, it is thoroughly mixed to precipitate the protein. The tubes were incubated at room temperature for 30 min and centrifuged for 15 min at 2500 rpm. To 0.4 ml of supernatant obtained by centrifugation, 0.4 ml of Ellman's Reagent (DTNB) was added. Then, to make the volume up to 3 ml, phosphate buffer (0.1 M) was added to the samples. On reaction with Ellman's reagent, reduced GSH produces a compound in yellow color. The absorbance of colored compound is read at 412 nm against the blank reagent. The unit of the result is expresses in µmole/min/mg protein [8].

**Lipid peroxidation estimation**

By estimating the quantity of malondialdehyde (MDA) formed, the oxidative stress extension can be quantitatively detected since MDA is an end product of lipid peroxidation. This is considered as a biomarker in evaluating the toxic stress levels in cells. The protocol followed for performing this assay was according to the method of Ohkawa et al. [9]. 1 ml of 10 % trichloroacetic acid (TCA) was added to equal volume of homogenate kept at room temperature for 10 minutes. The mixture was centrifuged at 2500 rpm for 15 minutes and 1 ml of supernatant obtained was added to 1 ml of TCA (0.67%), mixed, and kept for 20 minutes in boiling water bath. Then, the tubes were kept under running tap water to cool and 5 ml of n-butanol: Pyridine and distilled water (1:1) were added to the tubes. Again, after centrifuging the mixture, the supernatant of butanol layer obtained was separated, and the absorbance was read at 532 nm. The results were expressed in nmole MDA formed/mg protein.

**SOD assay**

The enzyme SOD acts as a defensive agent against some highly reactive free radicals such as hydroxyl radicals by converting it into oxygen and hydrogen peroxide. To determine SOD activity, Marklund's pyrogallol autoxidation method was followed [10]. 0.5 ml of supernatant from homogenate was mixed in 0.15 ml of chloroform and 0.25 ml absolute ethanol. For 15 minutes, the tubes were left in shaker followed by centrifugation. To the supernatant of 0.5 ml, 2 ml of 0.1 M Tris HCl with pH 8.2 was mixed along with 0.5 ml of pyrogallol solution and distilled water of 1 ml. With a time interval of 1 minute between each reading, the absorbance was read at 420 nm for 3 minutes. The results of the enzymatic activity were expressed in U/mg protein.

**CAT activity estimation**

Hydrogen peroxide formed from toxic activities is able to get converted to oxygen and water with the action of enzyme CAT. The reaction mixture tubes with 1 ml phosphate buffer (0.01 M and pH 7.0), 0.5 ml of 0.2 M H₂O₂, 0.4 ml distilled water, and 0.1 ml of supernatant from homogenate were incubated for 1 minute at room temperature. Potassium dichromate: Glacial acetic acid solution was added to the tubes followed by putting in boiling water bath for 15 minutes. Tubes are allowed to cool under tap water and the absorbance reading was taken at 570 nm. The results were expressed in µmole of H₂O₂ consumed/minutes mg protein [11].

**Histological examination**

Liver samples excised from each group were placed in 10 % neutral buffered solution of formalin. The liver tissue slices were then processed using an automated tissue processor followed by embedding in wax. The sections were cut out with 5 µm thicknesses using Leica microtome RM 2155 and stained the samples using Hematoxylin and Eosin (H & E). The stained slides were then observed under the light microscope.

**Statistical analysis**

The statistical study was done using the ANOVA method followed with Dunnett's test, and all the data were expressed in means±standard error mean. Statistically, p<0.05 was considered statistically significant. It was carried out utilizing the software Graphpad Instat software Inc., Version 3.06, San Diego, USA.

**RESULTS**

**Body and organ weight**

The gain in body weight, changes, and variations in liver weight on the administration of lead acetate and pretreatment effect of *C. quadrangularis* on lead acetate intoxicated rats are given in Table 1. Body weight got reduced significantly in Group II when compared to Group I. However, the body weight got increased in Groups III, IV, and V by 1.05%, 14.44%, and 15.74%, respectively, when compared to the Group II (negative).

**Biochemical estimation**

Due to liver damage, the lipid peroxidation levels are meant to get highly elevated, and it was clearly observed in Group II (negative) when compared to Group I. The gain in body weight, changes, and variations in liver weight on the administration of lead acetate and pretreatment effect of *C. quadrangularis* on lead acetate intoxicated rats are given in Table 2. Increased levels of marker enzymes ALP and AST were observed in Group II (negative) when compared with the control Group I indicating liver impairment. Groups III, IV, and V exhibited decreased levels of the marker enzymes implying that the lead acetate activity got counteraction by *C. quadrangularis* administration. Serum total protein levels were also reduced significantly in Group II when compared with Group I and the groups pretreated with *C. quadrangularis* also exhibited values close to the normal value and significant increase in comparing with Group II.

**Antioxidant assays**

Table 3 depicts the effect of *C. quadrangularis* over the actions of various antioxidant enzymes like SOD, CAT, etc. In case of liver damage, the lipid peroxidation levels are meant to get highly elevated, and it was clearly observed in Group II (negative) when compared to Group I, and in the groups pretreated with *C. quadrangularis*, the lipid peroxidation levels were significantly reduced when compared to Group II (negative). Reduced GSH level was also reduced significantly in Group II in comparison with Group I. The gain in body weight, changes, and variations in liver weight on the administration of lead acetate and pretreatment effect of *C. quadrangularis* on lead acetate intoxicated rats are given in Table 3 also depicts the effect of *C. quadrangularis* over the actions of some antioxidant enzymes such as SOD and CAT. Levels of these enzymes were decreased significantly in Group II (negative) when compared to Group I (control).
In other three groups, the levels of these enzymes were restored back to normal when compared with Group II. It indicates the protective effect of *C. quadrangularis*.

**Histopathological examinations**

Fig. 1 exhibits the histological variations on the effects of lead and pre-treatment with *C. quadrangularis*. Group I, normal control (Fig. 1a) depicts a normal histology, but Group II, lead-acetate-induced group (Fig. 1b) shows damaged histology, characterized by congested, and dilated central vein and hepatocytes showing necrosis, karyolysis, and pyknosis of nuclei which are indications of hepatocellular injury. Chronic inflammatory cell infiltration in the portal tract with destruction of the perportal hepatocytes was observed in the Group III (Fig. 1c). In Group IV, minimal with congested and dilated central veins and congestion of vessels in portal tract with minimal inflammation with more of normal hepatocytes were observed (Fig. 1d). In group V, almost normal liver architecture with minimal inflammation in the portal tract and normal hepatocytes (Fig. 1e) were observed.

**DISCUSSION**

Liver being the major site of xenobiotic metabolism can exhibit the elevated quantity of toxic substances which may lead to impairment in liver function by changing the antioxidant reserve [12]. There are several causes responsible for liver damage for which lead accumulation in the body is also a reason. ALP and AST are considered as well-known markers of hepatic injuries. Elevated levels of these enzymes in the serum of lead acetate treated animals indicate hepatic damage as they will flow out from the liver, because of tissue deterioration, to the blood. The antioxidant defense depletion or increase in production of free radicals will result in deterioration of antioxidant-prooxidant balance, which later leads to oxidative stress-induced apoptosis [13]. Pre-treatment with *C. quadrangularis* could prevent lipid peroxidation either directly or indirectly by free radical scavenging through GSH. Under oxidative stresses, GSH will get changed to GSH disulfide (GSSG), and its depletion leads in lipid peroxidation. Hence, GSH can be used as a marker for checking the oxidative stress levels. Declined levels of GSH in lead acetate treated Group II are may be due to the increased utilization of GSH. Pre-treatment using methanolic extract of *C. quadrangularis* resulted in restoring the GSH levels. It can be due to the fundamental reduction in hepatic peroxidative actions leading to restoration of GSH. SOD and CAT act as a guarding system against ROS [12,14]. In this study, SOD levels were drastically decreased in lead-acetate-induced Group II animals. The decreased levels of these enzymes can be due to the fact that inactivation of H$_2$O$_2$ scavenging enzymes resulting from the SOD inactivation by superoxide anions. SOD and CAT are mutual supporting teams of the antioxidant enzyme. Pre-treatment with methanolic extract of *C. quadrangularis* could prevent depletion in the actions of SOD and CAT effectively by scavenging on the radicals responsible for the depletion of these enzymes. The anabolic steroid isolated from *C. quadrangularis* found to act on estrogen receptors of bone cells.

**CONCLUSION**

*C. quadrangularis* has hepatoprotective activity against lead acetate-induced liver injury. The dominant mechanisms involved in the hepatoprotective activity of *C. quadrangularis* consist of restoration of the tissue level SOD, GSH, and CAT enzyme levels implying the restoration of inherent protection apparatus by various phytochemical constituents present in *C. quadrangularis*. However, further detailed studies toward clinical and molecular levels are required to understand the exact action mechanism.

### Table 1: Effect of lead acetate administration on body and organ weight with or without the administration of *C. quadrangularis*

| Groups          | Initial body weight | Final body weight | Absolute organ weight | Percentage increased |
|-----------------|---------------------|-------------------|-----------------------|---------------------|
| Group I (control) | 176.3±4.25          | 208.6±6.29        | 4.36±0.27             | 15.4±1.77           |
| Group II (negative) | 177.5±3.09          | 203.1±3.07        | 3.31±0.21             | 1.48±2.49           |
| Group III (low dosage) | 170.3±5.83          | 203.1±3.07        | 4.13±0.29             | 15.4±1.77           |
| Group IV (medium dosage) | 173.3±4.96          | 216.1±5.83        | 4.25±0.27             | 15.4±1.77           |
| Group V (high dosage) | 180.1±3.91          | 214.6±5.32        | 4.36±0.23             | 15.7±1.47           |

Values are mean±SEM n=6. *p<0.05 and comparisons were as follows: *Group I versus Groups II, III, IV; *Group II versus Groups III, IV, V. *C. quadrangularis: Cissus quadrangularis*

### Table 2: Effect of *Cissus quadrangularis* on serum total protein and other liver function markers over lead acetate administration

| Groups          | AST (IU/L) | ALP (IU/L) | Total protein (g/dL) |
|-----------------|------------|------------|---------------------|
| Group I (control) | 74.7±5.03  | 177.5±4.95 | 10.4±0.22           |
| Group II (negative) | 90.9±3.9** | 330.7±7.8* | 9.2±0.09*           |
| Group III (low dosage) | 81.9±3.5** | 300.5±4.20** | 9.7±0.08*           |
| Group IV (medium dosage) | 76.5±3.18** | 261.7±17.2** | 10.3±0.09*           |
| Group V (high dosage) | 72.5±3.52** | 216.4±9.21** | 11.1±0.07*           |

Values are mean±SEM n=6. *p<0.05 and comparisons were as follows: *Group I versus Groups II, III, IV, V; *Group II versus Groups III, IV, V. *C. quadrangularis: Cissus quadrangularis; AST: Aspartate transaminase, ALP: Alkaline phosphatase*

### Table 3: Lead acetate effect on antioxidant enzymes in control and the experimental animals (with or without the prior administration of *Cissus quadrangularis*)

| Groups          | Group I (control) | Group II (negative) | Group III (low dosage) | Group IV (medium dosage) | Group V (high dosage) |
|-----------------|-------------------|---------------------|------------------------|--------------------------|----------------------|
| SOD (U/mg protein) | 13.8±0.61         | 8.18±0.49**         | 8.82±0.13              | 9.18±0.15**              | 9.7±0.27**           |
| CAT (umole of H$_2$O$_2$ consumed/min mg protein) | 80.5±0.54         | 52.20±0.71**        | 64.78±0.72**           | 73.39±1.12**            | 75.39±1.17**         |
| GSH peroxidase (µg of GSH utilized/min mg protein) | 4.20±0.14         | 27.88±0.43**        | 32.91±0.97**           | 35.25±0.14**            | 37.81±0.80**         |
| Lipid peroxidation (n moles/mg protein) | 6.06±0.62         | 10.77±0.52**        | 9.47±0.18**            | 7.05±0.46**             | 6.14±0.39**          |

Values are mean±SEM n=6. *p<0.05 and comparisons were as follows: *Group I versus Groups II, III, IV, V; *Group II versus Groups III, IV, V. *C. quadrangularis: Cissus quadrangularis; GSH: Glutathione, SOD: Superoxide dismutase, CAT: Catalase*

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