Groundwater Contaminant Identification and Modelling Using Principal Component Analysis and Geostatistical Techniques

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Abstract
Groundwater which constitutes high percent of the global fresh water is the most important source of drinking water, which when polluted, have acute effects on its users. Consequently, the quality and pollution of groundwater is a health concern in the world. The target of this research is to evaluate the quality of groundwater around the Niger Delta Basin Development Authority in order to identify and analyze the distribution of the critical contaminants that affect the overall quality of groundwater around the study area. About hundred (100) boreholes spread to cover the study area were sampled. The water samples were analyzed using standard procedures for assessing drinking water qualities in order to understand the existing condition of groundwater within the study area. Statistical analysis of the groundwater quality data was done using average weighted index method to compute the water quality index, factor analysis using principal component method to identify the groundwater contaminants affecting the overall groundwater quality and geospatial analysis using kriging interpolation method to evaluate the spatial distribution of the selected groundwater contaminants. From the principal component analysis, result revealed that; nitrate, total dissolved solids, concentration of iron, total suspended solids and turbidity were the most important contaminants affecting the quality of the groundwater. Result of geospatial analysis using kriging interpolation revealed that; the water quality parameters showed relatively strong degree of spatial dependency which made it possible to generate the spatial distribution map for the selected water quality parameters.

Keywords: Geospatial analysis, Kriging interpolation, water quality index, Principal component analysis.
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1. Introduction
The availability of water in adequate quality and quantity is essential for the existence of human life. Water is one of the basic elements that sustain life on earth. Man can survive for a considerable amount of time without food, but he will succumb within days without water. It is reported in (Nichole, 2004; Akporido, 2000) and Water for Life, (2007) that where there is water, there is life; and that life exists around numerous uses of water which make it important for survival and luxury. The average daily intake of water per person, directly or as contained in other foods, is about four litres (Water for life (2007). For those who live in hot dry climate, this daily requirement is at least double (Akporido, 2000; Water for Life, 2007). Whenever a number of people live together, a supply of water is always paramount. On this note, for all purposes for which water is required, the quality of water is of great importance (Ocheri et al., 2014). As world population is constantly growing, the demand for water increases every day.

Groundwater is one of the major sources of water. Groundwater is water stored under the earth surface in rocks and soil segments called aquifers and is less susceptible to contamination by the action of microorganism (Debels, et al., 2005). As groundwater use has increased, issue associated with the quality of groundwater resources have likewise grown in importance. Groundwater, depending on the depth and the prevailing soil condition is vulnerable to pollution by anthropogenic activities, such as agricultural sources, urbanization, un-engineered disposal of wastes including de-icing salt, oil and chemical spillage (Baalousha, 2010). These sundry activities are carried out by man without due regard to its consequences on the long run. Groundwater, once polluted can be very difficult to remediate back to its natural pristine state. For many years, attention has been directed at contamination from point sources. More recently, concerns have increased about nonpoint sources of contaminant and about the overall quality of groundwater resources (Faisal et al., 2003; Efe, 2003). Groundwater contamination can be detected by analyzing borehole water for a series of dissolved ion species. Hence water quality data sets typically contain many variables measured at several spatially scattered locations.

A review conducted in Ocheri, (2014), on groundwater quality in Nigeria urban areas revealed that Nigerian urban groundwater quality is influenced by the geological condition of the soil through which it flows, geochemical and physical process of the environment, rate of urbanization, industrialization and seasonal variability. Abams et al, (2007) and Agbalagba et al, (2011) conducted a review on the quality of groundwater in parts of Yenagoa and Port Harcourt and reported that the groundwater in these areas have been negatively impacted by anthropogenic activities in the areas.
2. Materials and Methods

2.1 Description of study area

The study area for this research is the Niger Delta Basin Development Authority. This study covers the original area of operation of the River Basin Authority, which is Rivers and Bayelsa State alone. The geographical coordinates of Rivers and Bayelsa states are 4.8581°N and 6.9209°E and 4.25°S and 5.37°W and 6.75°E respectively (Nwankwoala et al., 2011). The Niger Delta Basin is situated in the south-south geo-political zone of Nigeria. It is located in the rain forest region with relative humidity above 80% having an annual temperature range of 25°C to 31°C and annual rainfall of 4700mm on the coast to about 2400mm. The basin is characterized by two alternating climatic conditions of a long period of rainy season spanning from March to November, followed by a dry season spreading from November to March (Nwankwoala, et al., 2011). Figures 1 and 2 shows the Google earth and the study area maps respectively.

![Figure 1: Google earth map of study area (Google .com)](image1)

![Figure 2: Map of study area ((Bolaji and Tse, 2009)](image2)

2.1.1 Geology and Hydrogeology of study area

The Niger Delta Basin is located on the continental margin of the Gulf of Guinea in equatorial West Africa. The Niger Delta lithofacies is made up of the three distinct vertical subdivisions viz. the Benin formation, the Agbada formation and the Akata formation. The Benin formation being the upper delta-top Lithofacies comprises of massive continental sands and gravels. The Agbada formation or facie consists of the pro-delta marine shales, with low stand turbidite fans which are deposited in a deep marine setting. In the Northern Delta Sector during the Oligocene times the Benin formation first occurs (Bolaji and Tse, 2009). Similarly, Paleocene age was established as the occurrence of the Akata formation in the proximal parts of the Delta. The Niger Delta complex geomorphologic features comprise of fresh water swamps, mangrove swamps, beaches, bars, and estuaries (Bolaji and Tse, 2009).
2.2 Sampling location and sample collection

The boundary of built up area (land use) within the study area was digitized and gridded at 2km interval to determine the sampling points and ensure uniform coverage. Water samples was collected systematically so as to have a general overview of the water quality condition within the study area. For accurate geo-referencing of the selected boreholes, Garmin hand held GPS receiver was employed to determine the geographical coordinates of each borehole. A section of the boreholes sampled including their location and geographical coordinates is presented in Table 1. One hundred (100) boreholes were systematically sampled with reference to location points at each season: Wet season (July to October 2018) and dry season (November to December 2018) in order to determine the physico-chemical and biological parameters of the groundwater samples. At every point of collection, the air tight, clean and dried plastic containers were rinsed two to three times with the borehole water to be sampled before collection. The samples were labelled properly and stored in air tight, clean and dried plastic containers before been transported to Water Resources and Environmental laboratory in the Department of Civil Engineering, University of Benin where the analysis were conducted in line with standard procedures and guideline recommended by World Health Organization (WHO). The water samples were analyzed in triplicates to obtain the mean value and standard deviation of each water quality test parameters. For the analysis of biochemical oxygen demand (BOD), the black bottles containing the water samples remained tightly closed prior to analysis in order to prevent photosynthetic and oxygen generation. In-situ parameters, namely; dissolved oxygen (DO), temperature, pH electrical conductivity (EC) and total dissolved solids (TDS) were determined in the field immediately after sample collection to avoid false measurement values (APHA, 2005).

Table 1: Coordinate Data of Sampled Boreholes (RIVERS)

| Borehole Codes | Locations                  | Northing | Easting |
|----------------|----------------------------|----------|---------|
| 1. Sample R1  | Igbu Ahaoda                | 239820   | 561471  |
| 2. Sample R2  | Mini Ama                   | 269110   | 525361  |
| 3. Sample R3  | Arukuwo-Abua               | 235669   | 537656  |
| 4. Sample R4  | Bakana                     | 286341   | 528043  |
| 5. Sample R5  | Edeoha-Ahoada              | 237214   | 556600  |
| 6. Sample R6  | Edeoha-Ahoada              | 236203   | 556600  |
| 7. Sample R7  | Okoboh-Abua                | 235766   | 540433  |
| 8. Sample R8  | Buguma                     | 262207   | 524264  |
| 9. Sample R9  | Air force Base             | 280557   | 534103  |
| 10. Sample R10| Trans Amadi                | 279389   | 530030  |
| 11. Sample R11| Ipo-Ikwerre                | 274121   | 532098  |
| 12. Sample R12| Woji                      | 286716   | 533642  |
| 13. Sample R13| Rumuokwurushi (1)          | 283293   | 536010  |
| 14. Sample R14| Amakiri Polo               | 286238   | 527163  |
| 15. Sample R15| Rukpokwu                   | 289003   | 534162  |
| 16. Sample R16| Aggrey                    | 280451   | 526634  |
| 17. Sample R17| NDBDA                     | 278741   | 529397  |
| 18. Sample R18| Rumuokwurushi (2)          | 283012   | 536068  |
| 19. Sample R19| Amadi-Ama                 | 279849   | 530118  |
| 20. Sample R20| Owodu                     | 287302   | 531219  |
| 21. Sample R21| Okochiri                  | 307314   | 519241  |
| 22. Sample R22| Trans Amadi (3)            | 278023   | 530112  |
| 23. Sample R23| Railway                   | 279801   | 527029  |
| 24. Sample R24| Bundu                     | 279684   | 525973  |
| 25. Sample R25| Oyorokoto                 | 325714   | 496236  |
| 26. Sample R26| Kono Town                 | 334047   | 508598  |
| 27. Sample R27| Oyigbo (1)                | 289245   | 538032  |
| 28. Sample R28| Ngo Town Andoni           | 323819   | 495804  |
| 29. Sample R29| Yegha Gokona              | 319044   | 517018  |
| 30. Sample R30| Oyigbo (2)                | 289599   | 538240  |
| 31. Sample R31| Nyokuruk                  | 339050   | 510170  |
| 32. Sample R32| Tegu-Gokana               | 316831   | 519746  |
| 33. Sample R33| Woji (2)                  | 286421   | 533116  |

2.3 Water Quality Analysis

A total of thirty-three (33) physico-chemical parameters and two (2) microbiological parameters were analyzed for each sampled domestic borehole to provide an insight into the overall quality of water within the study area. The physico-chemical parameters include: temperature, odour, colour/clarity, total hydrocarbon content (THC),
pH, Electrical conductivity (EC), Turbidity, Total suspended solid (TSS), Salinity, Alkalinity, Total Dissolved Solids (TDS), and Dissolved Oxygen (DO). Others are; Biochemical Oxygen Demand (BOD), Chemical Oxygen Demand (COD), Bicarbonate (HCO$_3$), Sodium (Na), Potassium (K), Calcium (Ca), Magnesium (Mg), Chloride (Cl$^-$), Phosphorus (P), Ammonium (NH$_4$), Nitrite (NO$_2$), Nitrate (NO$_3$), Sulphate (SO$_4$) and heavy metals, namely; Iron (Fe), Manganese (Mn), Zinc (Zn), Copper (Cu), Chromium (Cr), Cadmium (Cd), Nickel (Ni) and Lead (Pb). The microbiological parameters include: Total Coliform Counts (TCC) and E. Coli

### 2.3.1 Determination of in-situ parameters; (pH, EC, TDS, DO and Temperature)

For electrical conductivity (EC), total dissolved solids (TDS), pH, temperature and dissolved oxygen (DO), in-situ measurements were carried out since the measurement values of the parameter’s changes with storage time (WHO, 2003). pH, electrical conductivity, temperature and total dissolved solids were measured using portable meter’s (multi-parameters) while dissolved oxygen was examined using DO meter (Lutron DO-5509, Range 0 – 20mg/l) shown in Figure 3

![DO meter and multi portable meter](image)

The multi portable meter probe was submerged in the water at 4cm and pH mode selected. Water sample was stirred gently and pH value displayed on the meter was allowed to adjust and stabilize before recording. Other measurements buttons were pressed successively and values recorded. The procedure was repeated three (3) times and the mean value calculated for each parameter. DO meter was also inserted into the water sample at about 10cm depth using the oxygen probe handle.

UNICAM 969 Atomic Absorption Spectrometer (AAS) shown in Figure 4 was used to determine the concentration of heavy metals such as; Iron (Fe), Manganese (Mn), Zinc (Zn), Copper (Cu), Chromium (Cr), Cadmium (Cd), Nickel (Ni), Lead (Pb), and Vanadium (V) while UV visible spectrophotometer (Thermo Scientific Spectronic 20D+) presented in Figure 5 was used to analyzed the level of phosphorous (P), Nitrate (NO$_3$), Nitrite (NO$_2$) and Sulphate (SO$_4$). Other apparatus utilized included 250ml separating glass funnels, Cuvette, 10ml and 50ml pipette, 250ml conical flask, 50ml burette, 25ml and 50ml volumetric flask, glass beads, refrigerator, oven and whatman filter paper.
Preparation of reagents and procedures employed in the laboratory for the analysis and determination of all water quality parameters followed the standard methods recommended by relevant authorities such as World Health Organization (WHO).

2.4 Groundwater contaminant identification

Statistical computations involving the application of principal component analysis for groundwater analysis was done using statistical package for the social sciences (SPSS 22 software). Principal component analysis (PCA) is a powerful tool that attempts to explain the variance of a large dataset of intercorrelated variables with a smaller set of independent variables. The Principal Components (PC) is the uncorrelated (orthogonal) variables obtained by multiplying the original correlated variables with the eigenvector, which is a list of coefficients (loadings or weightings). Thus, the PCs are weighted linear combinations of the original variables. PC provides information on the most meaningful parameters, which describe the whole data set while affording data reduction with a minimum loss of original information (Gajbhiye et al., 2014). Application of PCA to ground water analysis was conducted by using the following step by step methodology

i. Computation of Anti-image correlation matrix
ii. Computation of eigen values
iii. Extraction of the component matrix
iv. Extraction of the rotated component matrix
v. Horizontal decentralization of the component matrix

2.4.1 Anti- Image Correlation Matrix

The anti-image correlation matrix was employed to test the suitability of principal component analysis (PCA) in explaining the underlying correlation between selected water quality parameters. The suitability of PCA is based on the values of the off-diagonal matrix. If the off-diagonal matrix are very close to zero, then PCA is suitable otherwise we must think of another statistical method (Gajbhiye et al., 2014).

2.4.2 Computation of Eigen values

The eigen value is normally used as cutoff in factor analysis since it is the sum of the squared factor loadings of all variables. Factors with eigen values greater than 1 represent the number of component factors needed to describe the underlying variation of the groundwater quality. This are the component factors that contributes an adequate amount to the variation in the ground water quality. Factors with eigen value less than 1 means that such factor do not have any influence on the overall groundwater quality.

2.4.3 Computation of communalities

It is also important in the extraction phase to examine the communalities. The communalities is represented by the sum of the square loading for a variable across factor. Communality can range from 0 to 1. Communality of 1 means that all of the variation in the groundwater quality is explained by the component factors.

2.4.4 Extraction of component matrix

In order to identify the groundwater quality parameters that make up the members of each component factors, the component matrix was generated. The extracted component matrix was employed to understand the correlation between the groundwater quality parameters in each component group. In regression terms, the component matrix is the standardized regression coefficient between the observed values and the component factors. Higher factor loading indicates that a parameter is closely associated with the component factor. To determine the water quality parameters that make up each of the component factors, horizontal decentralization of the component matrix was done and the best favoured parameters was selected as member of that particular component factor.
2.4.5 Extraction of the rotated component matrix

The rotated component matrix gives you the factor (s) you need to pay more attention to in order to improve the overall quality of the groundwater in any location. To understand the factors that is most critical towards improving the overall groundwater quality of the study area, vertical decentralization of the rotated matrix was done (Gajbhiye et al., 2014). For principal component analysis (PCA) to work, the mean was subtracted from each of the data dimensions. The mean subtracted represent the average across each dimension. The mean was computed from

\[ \bar{X} = \frac{\sum_{i=1}^{n} X_i}{n} \]  

(1)

To compute the covariance matrix, equation (2) was employed as follows;

\[ Var(X) = \frac{\sum_{i=1}^{n} (X_i - \bar{X})(X_i - \bar{X})}{n - 1} \]  

(2)

From equation (2), the covariance matrix was computed using

\[ Cov(X, Y) = \frac{\sum_{i=1}^{n} (X_i - \bar{X})(Y_i - \bar{Y})}{n - 1} \]  

(3)

Covariance is normally measured in 2-dimensions such as (X, Y); (X, Z) or (Y, X). Provided the X and Y covariance matrix are of the same size, addition and multiplication of the covariance matrix was employed to calculate the eigenvector and eigenvalues of the covariance matrix. The notion of data compression and reduced dimensionality comes into PCA in computing the feature vector. To compute the feature vector, equation (4) was employed as follows

Feature Vector = (eig1, eig2, eig3 ------------------ eign)  

(4)

Once the component eigenvector is chosen, the feature vector was formed and then transposed to get the final solution as follows; Final Data = (Row Feature Vector) * (Row Data Adjust)  

(5)

2.5 Groundwater analysis using geostatistical techniques

The geostatistical analyst is an Arcmap extension that provides a wide variety of tools for spatial data exploration, identification of data anomalies and evaluation of error in prediction surface models, statistical estimation, and optimal surface creation. The geostatistical analyst can determine the approximate concentration of selected water quality parameters in an unsampled location and can determine the spatial distribution of the parameter i.e. where they may be moving by creating an optimal interpolated surface.

The step by step methodology involved in the creation of interpolation surface for selected groundwater quality parameters are:

i. Collection of water sample from selected boreholes within the study area
ii. Collection of spatial data (Northings, Easting and Elevation) of the selected boreholes within the study area using Garmin Hand-held GPS
iii. Analysis of the physico-chemical and microbiological properties of the collected groundwater samples
iv. Modelling the spatial distribution of selected groundwater parameters using kriging interpolation in ArcGIS

Kriging is a geostatistical interpolation technique that considers both the distance and the degree of variation between known data points when estimating values in unknown areas. It attempts to minimize the error variance and set the mean of the prediction errors to zero so that there are no over- or under-estimates. It is a robust interpolation tool which derives weights from surrounding measured values to predict values at unmeasured locations. The kriging weights are obtained from fitting of semi- variogram models, developed by viewing the spatial structure of the data. To create a continuous surface or map of any phenomenon, predictions are made from the models and the spatial arrangement of measured values. In this study, ordinary kriging method was employed for the spatial analysis of some selected groundwater quality parameters owing to its simplicity and prediction accuracy in comparison to other kriging methods. In addition, it is an error minimization technique where the values of a property (water quality) at unsampled location are predicted such that the variance of the estimated value is reduced to the barest minimum. The following steps are involved in the use of kriging interpolation method for the geospatial analysis of selected groundwater quality parameters

i. Evaluation of normality test
ii. Selection of attribute data and model interpolation method
iii. Semivariogram fitting and testing
iv. Cross validation
v. Spatial Dependency determination
vi. Creation of groundwater quality prediction maps

2.5.1 Test of Normality
To test the normality of the water quality datasets, histogram plots of the different water quality parameters were generated using statistical package for the social sciences (SPSS version 22) software

2.5.2 Selection of attribute data and model interpolation method
Geostatistics can reveal stochastic structure of groundwater quality variations in space and can showed a significant variation in the pH of groundwater within and around the study area. Table 2 presents the modelling step for the spatial variation of pH and other selected water quality parameters. It is observed that the attribute data selected is pH while the interpolation method is Kriging/Cokriging method.

Table 2: Kriging modeling for the spatial variation of pH

The kriging interpolation method was selected based on its capability to determine the corresponding value of the attribute data in an unsampled location. The kriging method is also very flexible and allows you to investigate the graph of auto-and cross-correlation. In addition, the kriging method uses statistical models that allow a variety of output surfaces including predictions, prediction standard errors, probability and quantile. More also, unlike other methods of interpolation that assumes that the input data are normally distributed, kriging interpolation method recognizes the stochastic nature of the input data.

2.5.3 Fitting and testing of semivariogram
The semivariogram/covariance model allows you to select the model that best described the groundwater quality data. In addition, the nugget, range and partial sill can also be determined. The interphase of Arcmap showing the semivariogram/covariance modelling of pH is presented in Figure 6

Figure 6: Semivariogram/covariance modelling of pH
2.5.4 Estimation of cross validation statistics
The Cross Validation statistics gives you an idea of how well the model predicts the values at the unknown locations. The essence of cross validation was to examine closely the fitted models and select the model that gives the best result (prediction). To select the model that gives the best result, goodness of fit statistics, namely; Root mean square error (RMSE), Mean square error (MSE), Root mean square standard error (RMSSE) and Average standard error (ASE) were employed

3. Results and Discussion
Principal component analysis (PCA) was performed to identify the most critical water quality parameter(s) that significantly influenced the overall quality of groundwater within the study area. To perform the analysis, water samples were collected from hundred (100) boreholes and twenty one (21) water quality parameters were monitored during dry and wet season. These water quality parameters include; pH, Electrical conductivity (EC), Salinity, Total Dissolve Solids (TDS), Dissolved Oxygen (DO), Bicarbonate (HCO$_3$), Sodium (Na), Potassium(K), Calcium (Ca), Magnesium (Mg), Chloride (Cl$^-$), Phosphate (PO$_4^{3-}$), Nitrate (NO$_3$), Sulphate (SO$_4^{2-}$), Iron (Fe), Zinc (Zn), Copper (Cu), Turbidity, Total suspended solid (TSS), Temperature and Alkalinity.

3.1 Testing the suitability of PCA
To ascertain the potential of principal component method in handling the specific objectives of this study, Kaiser-Meyer-Olkin measure of sampling adequacy and anti-image correlation matrix were employed. The estimated statistics based on Kaiser-Meyer-Olkin measure of sampling adequacy is presented in Table 3

Table 3: Testing the suitability of principal component analysis

| KMO and Bartlett's Test | Kaiser-Meyer-Olkin Measure of Sampling Adequacy | Bartlett's Test of Sphericity |
|-------------------------|-----------------------------------------------|------------------------------|
|                         |                                               | Approx. Chi-Square df Sig.   |
|                         | .023                                          | 2.93E3 0.000                 |

From the result of Table 3, it was observed that the significant value based on Bartlett test of sphericity was 0.000. Since the calculated (p-value) was less than 0.05, it was concluded that the model is significant and that principal component analysis (PCA) was suitable for the analysis. A further test of suitability of principal component analysis is the use of anti-image correlation matrix presented in Table 4

Table 4: Anti-image correlation matrix

| Correlation Matrix | pH | Nh4 | Electrical Conductivity | Turbidity | Dissolved Oxygen | Total Dissolved Solids | Sodium | Sulphate | Calcium |
|--------------------|----|-----|-------------------------|-----------|------------------|------------------------|--------|----------|---------|
| pH                 | 1.00 |     |                         |           |                  |                        |        |          |         |
| Nh4                |       | 1.00 |                         |           |                  |                        |        |          |         |
| Electrical Conductivity |       |       | 1.00                     |           |                  |                        |        |          |         |
| Turbidity          |       |       |                         | 1.00      |                  |                        |        |          |         |
| Dissolved Oxygen   |       |       |                         |           | 1.00              |                        |        |          |         |
| Total Dissolved Solids |       |       |                         |           |                  | 1.00                   |        |          |         |
| Sodium             |       |       |                         |           |                  |                        | 1.00   |          |         |
| Sulphate           |       |       |                         |           |                  |                        |        | 1.00     |         |
| Calcium            |       |       |                         |           |                  |                        |        |          | 1.00    |

Result of Table 4 shows that principal component analysis is suitable for this analysis. The suitability of PCA is based on the fact that; the off diagonal matrix are less than one with a host of them very close to zero. Since the off diagonal matrix are less than one, it was concluded that PCA is good for the study.

3.2 Extraction of communalities
It is also important in the extraction phase to examine the communalities. Communalities represents the sum of
the square loading for a variable across factor. Communalities can range from 0 to 1. A communality of 1 means that all of the variation in the groundwater quality is explained by the component factor. The computed communalities is presented in Table 5.

Table 5: Computed communalities of groundwater quality parameters

| Parameter          | Initial | Extraction |
|--------------------|---------|------------|
| pH                 | 1.000   | .598       |
| Nitrate            | 1.000   | .739       |
| Electrical Conductivity | 1.000 | .934       |
| Turbidity          | 1.000   | .714       |
| Dissolved Oxygen   | 1.000   | .774       |
| Total Dissolved Solids | 1.000 | .955       |
| Sodium             | 1.000   | .687       |
| Sulphate           | 1.000   | .795       |
| Zinc               | 1.000   | .655       |
| Copper             | 1.000   | .692       |
| Chloride           | 1.000   | .594       |
| Iron               | 1.000   | .652       |
| Carbonate          | 1.000   | .711       |
| Total Suspended Solids | 1.000 | .744       |
| Phosphates         | 1.000   | .595       |
| Temperature        | 1.000   | .642       |
| Alkalinity         | 1.000   | .625       |
| Salinity           | 1.000   | .329       |
| Magnesium          | 1.000   | .858       |
| Potassium          | 1.000   | .735       |
| Calcium            | 1.000   | .796       |

Extraction communalities are estimates of the variance in each variables accounted for by the factors. High extraction indicates that the extracted components represents the variables well. If any extraction is very low (< 0.3), then one may need to extract another component factor. Since none of the extraction is less than 0.3, it was concluded that the extracted components are good enough to explain the underlying variation associated with the quality of groundwater.

3.3 Extraction of Component Factors

Using the method of principal component analysis, result of the component factor extracted is presented in Table 6.

Table 6: Extraction of component factors using principal component analysis method (PCA)

| Component       | Total Variance Explained | Total % Variance | Cumulative % | Total % Variance | Cumulative % | Total % Variance | Cumulative % |
|-----------------|--------------------------|------------------|--------------|-----------------|--------------|-----------------|--------------|
| pH              | 1.000                    | 43.885           | 43.885       | 43.885          | 43.885       | 45.014          | 45.014       |
| Nitrate         | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Electrical Conductivity | 1.000   | 43.885 | 43.885 | 45.014 | 45.014 | 45.014 | 45.014 |
| Turbidity       | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Dissolved Oxygen | 1.000                | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Total Dissolved Solids | 1.000  | 43.885 | 45.014 | 45.014 | 45.014 | 45.014 | 45.014 |
| Sodium          | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Sulphate        | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Zinc            | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Copper          | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Chloride        | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Iron            | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Carbonate       | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Total Suspended Solids | 1.000  | 43.885 | 45.014 | 45.014 | 45.014 | 45.014 | 45.014 |
| Phosphates      | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Temperature     | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Alkalinity      | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Salinity        | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Magnesium       | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Potassium       | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |
| Calcium         | 1.000                    | 43.885           | 43.885       | 45.014          | 45.014       | 45.014          | 45.014       |

Extraction Method: Principal Component Analysis.
The extraction analysis determines how well the component factors explain the variation in the overall quality of groundwater from the hundred boreholes using the total variance explained as presented in Table 6. Factors with eigen values greater than one represent the number of component factors needed to describe the underlying variation of the ground water quality. This are the component factors that contributes an adequate amount to the variation in the ground water quality. The eigen value is normally used as cutoff in factor analysis since it is the sum of the squared factor loadings of all variables. Factors with eigen value less than one means that such factor do not have any positive influence on the overall groundwater quality. From the results of Table 6, it was observed that five component factors had eigen value greater than one. These component factors contributed to change in the quality of the groundwater around the study area. This claim was further supported using the scree plot presented in Figure 7.

From the scree plot of Figure 7, it was again observed that five component factors possess very strong influence on the overall quality of the groundwater around the study location.

### 3.4 Extraction of Component Matrix

In order to identify the groundwater quality parameters that make up the members of each component factors, the component matrix was generated as presented in Table 7.

#### Table 7: Component matrix showing the five component factors

| Component Matrix | Component 1 | Component 2 | Component 3 | Component 4 | Component 5 |
|------------------|-------------|-------------|-------------|-------------|-------------|
| pH               | -0.597      | -0.388      | -0.209      | -0.127      | -0.170      |
| Nitrate          | 0.878       | -0.111      | -0.040      | 0.825       | -0.049      |
| Electrical Conductivity | 0.651 | -0.733      | -0.236      | -0.111      | 0.113       |
| Turbidity        | -0.193      | 0.059       | 0.566       | 0.508       | 0.063       |
| Dissolved Oxygen | -0.808      | 0.070       | -0.104      | -0.803      | 0.012       |
| Total Dissolved Solids | 0.328 | -0.040      | -0.274      | -0.227      | -0.129      |
| Sodium           | 0.817       | -0.024      | -0.126      | -0.814      | -0.044      |
| Sulfate          | -0.898      | 0.069       | -0.012      | 0.819       | 0.011       |
| Total Solid      | -0.784      | -0.103      | -0.087      | 0.877       | -0.018      |
| Copper           | 0.808       | -0.041      | 0.080       | 0.804       | -0.117      |
| Chloride         | -0.715      | -0.131      | -0.186      | -0.846      | 0.005       |
| Iron             | -0.364      | 0.217       | 0.534       | -0.315      | 0.296       |
| Carbonate        | -0.759      | -0.172      | -0.204      | -0.245      | -0.072      |
| Total Suspended Solids | 0.118 | 0.265       | 0.036       | -0.418      | 0.026       |
| Phosphorus       | 0.727       | -0.010      | -0.174      | -0.404      | -0.052      |
| Temperature      | 0.418       | 0.068       | 0.240       | 0.454       | 0.447       |
| Alkalinity       | 0.773       | -0.004      | 0.057       | -0.137      | 0.041       |
| Hardness         | 0.265       | -0.128      | -0.186      | -0.803      | 0.451       |
| Magnesium        | -0.913      | -0.011      | 0.145       | -0.823      | 0.050       |
| Potassium        | -0.814      | -0.119      | 0.086       | -0.803      | 0.037       |
| Calcium          | 0.374       | 0.041       | 0.451       | -0.117      | 0.189       |

Extraction Method: Principal Component Analysis. 5 components extracted.
The extracted component matrix was employed to understand the correlation between groundwater quality parameters in each component group. In regression terms, the component matrix is the standardized regression coefficient between the observed values and the component factors. Higher factor loading indicates that a parameter is closely associated with the component factor. To determine the water quality parameters that is most highly correlated with the component factors, horizontal decentralization of the component matrix was done and the best favoured parameters was selected as member of that particular component factor. Result of the horizontal decentralization of the component matrix is presented in Table 8

Table 8; Result of horizontal decentralization of the component matrix

| S/No | Variable Code | Variable Name | 1     | 2     | 3     | 4     | 5     |
|------|---------------|---------------|-------|-------|-------|-------|-------|
| 1    | X1            | pH            | 0.368 |       |       |       |       |
| 2    | X2            | Nitrate       |       | 0.878 |       |       |       |
| 3    | X3            | EC            |       | 0.753 |       |       |       |
| 4    | X4            | Turbidity     |       |       | 0.589 |       |       |
| 5    | X5            | DO            |       |       | 0.104 |       |       |
| 6    | X6            | TDS           |       |       | 0.840 |       |       |
| 7    | X7            | Sodium        | 0.817 |       |       |       |       |
| 8    | X8            | Sulphate      | 0.886 |       |       |       |       |
| 9    | X9            | Zinc          | 0.794 |       |       |       |       |
| 10   | X10           | Copper        | 0.806 |       |       |       |       |
| 11   | X11           | Chloride      | 0.715 |       |       |       |       |
| 12   | X12           | Iron          | 0.534 |       |       |       |       |
| 13   | X13           | Carbonate     | 0.758 |       |       |       |       |
| 14   | X14           | TSS           |       | 0.396 |       |       |       |
| 15   | X15           | Phosphate     | 0.727 |       |       |       |       |
| 16   | X16           | Temperature   | 0.454 |       |       |       |       |
| 17   | X17           | Alkalinity    | 0.770 |       |       |       |       |
| 18   | X18           | Salinity      |       | 0.451 |       |       |       |
| 19   | X19           | Magnesium     | 0.913 |       |       |       |       |
| 20   | X20           | Potassium     | 0.842 |       |       |       |       |
| 21   | X21           | Calcium       | 0.656 |       |       |       |       |

Results of Table 8 revealed that:

i. The first component factor is most highly correlated with nitrate, sodium, sulphate, zinc, copper, chloride, carbonate, phosphate alkalinity, magnesium, potassium and calcium

ii. The second component factor is most highly correlated with pH, electrical conductivity and total suspended solids.

iii. The third component factor is most highly correlated with dissolved oxygen and concentration of iron

iv. The fourth component factor is most highly correlated with turbidity and temperature while

v. The fifth component factor is most highly correlated with salinity

3.5 Extraction of Rotated Component Matrix

The rotated component matrix is presented in Table 9
Table 9: Rotated component matrix

The rotated component matrix gives you the factor(s) you need to pay more attention to in order to improve the overall quality of the groundwater in any location. To understand the factors that are most critical towards improving the overall groundwater quality of the study area, vertical decentralization of the rotated matrix was done and result obtained is presented in Table 10.

Table 10: Result of vertical decentralization of the rotated component matrix

| Variables          | Component Factors |
|--------------------|-------------------|
| S/No               | 1 2 3 4 5         |
| X₁                 | pH               |
| X₂                 | Nitrate          | 0.859 |
| X₃                 | EC               |
| X₄                 | Turbidity        | 0.792 |
| X₅                 | DO               |
| X₆                 | TDS              | 0.971 |
| X₇                 | Sodium           |
| X₈                 | Sulphate         |
| X₉                 | Zinc             |
| X₁₀                | Copper           |
| X₁₁                | Chloride         |
| X₁₂                | Iron             | 0.795 |
| X₁₃                | Carbonate        |
| X₁₄                | TSS              | 0.830 |
| X₁₅                | Phosphate        |
| X₁₆                | Temperature      |
| X₁₇                | Alkalinity       |
| X₁₈                | Salinity         |
| X₁₉                | Magnesium        |
| X₂₀                | Potassium        |
| X₂₁                | Calcium          |

From the result of Table 10, it was observed that; nitrate, total dissolved solids, concentration of iron, total suspended solids and turbidity were the most important variables affecting the quality of the groundwater within the study area. The component plot in rotated space is presented in Figure 8.
From the component plot of Figure 8, it was observed that the combined isolated variables is X3 and X6 which represent electrical conductivity and total dissolved solids an indication that both variables are not only highly correlated, they also jointly contributes to variation in the quality of groundwater within the study area.

The final result of principal component analysis revealed that; nitrate, total dissolved solids, concentration of iron, total suspended solids and turbidity were the most important variables affecting the quality of the groundwater.

To study the spatial distribution of these water quality parameters around the study location, geospatial analysis was done using kriging method of interpolation. The input variables for spatial analysis include; coordinates of the sampled boreholes (Northing and Easting), and the attribute variables (concentration of iron, nitrate, total dissolved solid, total suspended solids and turbidity).

3.6 Exploring Data Distribution

Exploring the data helped examine how the data is distributed and also to examine the global trends associated with the data. Geostatistical analysis provides many data exploration tools that helped examine the distribution of observed water quality parameters. For this study, frequency histogram plots were done to verify the distribution of the critical water quality parameters identified by principal component analysis and results are presented in Figures 8a, 8b, 8c, 8d and 8e respectively.

Figure 8a: Histogram plot of Nitrate
Figure 8b: Histogram plot of Iron

Figure 8c: Histogram plot of total suspended solids

Figure 8d: Histogram plot of total dissolved solids
From the result of Table 8a, it was observed that; though the concentration of nitrate was below the maximum allowable limit, the highest concentration of nitrate was recorded in water samples collected from Mini-Ama, Bakana and Woji. From the result of Table 8b, it was observed that; the concentration of iron in some of the borehole sampled was higher than the maximum allowable limit. The highest concentration of iron was recorded in water samples collected from Owodu, Okochiri, Rumuokwurushi, Rukpokwu, Amakin-Polo, Air force base, Buguma and Amakiri-polo. From the result of Table 8c, it was observed that; though the concentration of total suspended solids was below the maximum allowable limit, the highest concentration of TSS was recorded in water samples collected from Mini-Ama, Bakana, Woji, Owodu, Okochiri, Rumuokwurushi, Rukpokwu, Amakin-Polo, Air force base, Buguma and Amakiri-polo. Others are; Edeoha-Ahoada, Okoboh-Abua, Trans-Amadi, Aggrey, Amadi-Ama and Owodu. From the result of Table 8d, it was observed that; though the concentration of total dissolved solids was below the maximum allowable limit, the highest concentration of TDS was recorded in water samples collected from Rumuokwurushi, Amadi-Ama, Owodu, Trans-Amadi, Oyigbo and Yegha-Gokona. From the result of Table 8e, it was observed that; though the turbidity of the water samples was below the maximum allowable limit, the highest turbidity was recorded in water samples collected from Mini-Ama, Bakana, and Woji. Owodu, Okochiri, Rumuokwurushi, Rukpokwu, Amakin-Polo, Air force base, Buguma and Amakiri-polo others are; Edeoha-Ahoada, Okoboh-Abua, Trans-Amadi, Aggrey, Amadi-Ama and Owodu.

The histogram of Figures 8a, 8b, 8c, 8d and 8e revealed that the parameters are not normally distributed. This further support the claim that seasonal variation exist among the water quality parameters. The results are in line with that of previous study conducted by (Hooshmand et al., 2011) in which two geostatistical methods, kriging and co-kriging, were applied to estimate chloride content and sodium adsorption ratio (SAR) of ground water in the Boukan area of Iran. The histogram plot generated for the chloride concentration around the study area did not also obey normality an indication that ground water quality parameters varies with depth, time and season. In addition, in the study by (Ahmadi et al., 2008) in which geostatistical methods, namely; kriging and co-kriging were applied to study the maximum, minimum and mean ground water depths of 39 wells, the ground water depths were all observed to be spatially correlated an indication that they are not normally distributed. The basic steps involved in the development of the spatial distribution map of the critical water quality parameters around the study location using kriging interpolation method is presented as foloows.

3.6.1: Fitting and testing of semivariogram

Eight semivariogram models, namely; Circular, Spherical, Exponential, Gaussian, Hole effect, K-Bessel, J-Bessel and Stable) were fitted for each of the five critical water quality parameters used for geostatistical analysis during wet season in order to select the best fitted model for the parameter. Table 11 shows the result of the semivariogram models for nitrate and the corresponding values of nugget (the variability in the field data that cannot be explained by distance between the observations), major range (represents the distance at which two observations are unrelated/independent) and sill (the semi-variance at which the leveling takes place. The end point of range is the beginning point of sill. The different between the sill and the nugget is called partial sill).
Table 11: Semivariogram models for Nitrate

| S/N | Model Type | Nugget   | Major Range | Partial Sill |
|-----|------------|----------|-------------|--------------|
| 1.  | Circular   | 25.88929 | 122473.6   | 58.77035     |
| 2.  | Spherical  | 24.03614 | 122473.6   | 44.43717     |
| 3.  | Exponential| 23.88726 | 122473.6   | 29.36212     |
| 4.  | Gaussian   | 25.88929 | 122473.6   | 58.77035     |
| 5.  | Hole effect| 25.73296 | 122473.6   | 30.75632     |
| 6.  | K-Bessel   | 25.81664 | 122473.6   | 54.20273     |
| 7.  | J-Bessel   | 25.74801 | 122473.6   | 33.8048      |
| 8.  | Stable     | 24.02447 | 122473.6   | 38.82232     |

Fitted Semivariogram models presented in Table 11 provided information about the range, nugget and partial sill (Model Parameters) which were used to measure the degree of spatial dependency of sampled borehole points as a result of distance between them. It also provides the input parameters that were utilized for the kriging interpolation. To select the model that best described each water quality parameters and which was employed to generate the final prediction map, selected goodness of fit statistics generated from the cross-validation step were employed.

3.6.2: Estimation of cross validation statistics

To select the most suitable model needed to generate the final prediction map, selected goodness of fit statistics, namely; Root mean square error (RMSE), Mean square error (MSE), Root mean square standardized error (RMSSE) and Average standard error (ASE) were employed and the calculated values corresponding to the different models is presented in Table 12.

Table 12: Result of cross validation statistics

| S/N | Model Type | RMSE         | MSE         | RMSSE       | ASE      |
|-----|------------|--------------|-------------|-------------|----------|
| 1.  | Circular   | 6.828        | -0.00610    | 1.245       | 5.369    |
| 2.  | Spherical  | 6.944        | -0.00292    | 1.255       | 5.408    |
| 3.  | Stable     | 4.949        | -0.000252   | 1.255       | 5.415    |
| 4.  | Exponential| 6.972        | -0.000067   | 1.241       | 5.496    |
| 5.  | Gaussian   | 6.828        | -0.00610    | 1.245       | 5.369    |
| 6.  | Hole effect| 6.858        | -0.00343    | 1.250       | 5.367    |
| 7.  | K-Bessel   | 6.842        | -0.00439    | 1.247       | 5.369    |
| 8.  | J-Bessel   | 6.860        | -0.00286    | 1.250       | 5.370    |

Following the application of different models on each water quality parameter, the errors were calculated using cross validation and the model that gives the best result was chosen. The criterion for selecting the best model is as follows:

i. The mean standardized error should be close to 0,
ii. The root mean square error and average standard error should be as small as possible and close to each other
iii. The root mean square standardized error should be close to 1.

Based on the above criterion, and examining the results of Table 12, the stable model was selected as the best fit model for each selected water quality test parameters.

3.6.3: Examination of spatial dependency

Fitted semivariogram chart can be used to examine the spatial dependence between the measured sample points. Based on the results of Tables 11 and 12, the stable model was selected as the best fit model and a summary Table was generated and presented in Table 13.

Table 13: Summary table for estimating spatial dependence

| S/N | Parameter | Best Model | Nugget   | Major Range | Partial Sill |
|-----|-----------|------------|----------|-------------|--------------|
| 1.  | Nitrate   | Stable     | 24.02447 | 122473.6    | 38.82232     |
| 2.  | Iron      | Stable     | 26.0645  | 10564.9     | 19.0956      |
| 3.  | TSS       | Stable     | 3.44782  | 33089.6     | 55.0934      |
| 4.  | TDS       | Stable     | 13.4667  | 9.04586     | 6.78023      |
| 5.  | Turbidity | Stable     | 1.0456   | 107783.3    | 32.9067      |

The sill (c) is the summation of nugget and partial sill while the ratio of Nugget to Sill \( \frac{C}{c} \) was employed to measure the degree of spatial structure (dependence) of a water quality parameter. If the ratio is less than 25%, the variable has strong partial dependence; between 25% and 75%, the variable has moderate spatial dependence, and greater than 75%, the variable shows only weak spatial dependence. The computed spatial dependence for the selected water quality parameters using the selected best fit model is presented in Table 14.
Table 14: Estimated spatial dependence of water quality parameters

| Parameter | Best Model | Nugget \( (C_n) \) | Partial Sill | Sill \( (C) \) | \( [C_n/C] \) | Degree of Spatial Dependency |
|-----------|------------|---------------------|--------------|--------------|----------------|-----------------------------|
| Nitrate   | Stable     | 24.02447            | 38.82232     | 62.84679     | 0.38227        | Strong                      |
| Iron      | Stable     | 26.0645             | 19.0956      | 45.1601      | 0.577158       | Strong                      |
| TSS       | Stable     | 3.44782             | 55.0934      | 58.54122     | 0.058896       | Strong                      |
| TDS       | Stable     | 13.4667             | 6.78023      | 20.24693     | 0.665123       | Strong                      |
| Turbidity | Stable     | 1.0456              | 32.9067      | 33.9523      | 0.030796       | Strong                      |

Results of Table 14 revealed that the water quality parameters showed relatively strong degree of spatial dependency which made it possible to generate the spatial distribution map for the selected water quality parameters.

3.6.4: Generation of water quality parameters distribution map

Finally the prediction map which can be employed to predict nitrate, concentration of iron, total suspended solids (TSS), total dissolved solids (TDS) and turbidity of unsampled boreholes within the study area was generated and presented in Figures 9a to 9e respectively.

Figure 9a: Final prediction map for Nitrate distribution

Figure 9b: Final prediction map for Iron distribution
It was observed from the prediction map of Figures 9a to 9e that areas with red colour codes represent higher
values of Nitrate, Iron, TSS, TDS and Turbidity while areas with blue colour codes represent lower values of Nitrate, Iron, TSS, TDS and Turbidity. To validate the result of this study, a comparison was made with the results of previous studies as follows; in a study by Ramesh and Elango (2013). The study was carried out to analyze the spatial variability of groundwater quality for the region using samples collected from twenty-three (23) boreholes. The parameters analyzed included pH, EC, TDS, TH, Calcium, Magnesium, Bicarbonate, Chloride and Sulphate which were compared with permissible limits. Spatial interpolation technique using IDW approach was used to delineate the distribution of water pollutants. Results of the study revealed that the stable and K-Bessel model was the best fit model which is similar to the results of this study.

Research work done by Balakrishnan et al. (2011) on the use of GIS to map groundwater quality variation in Gulbarga City Karnataka India shows that GIS is a very powerful tool for investigating groundwater quality information. The study sampled 76(seventy-six) bore wells and open wells representing the entire corporation area. The analysis was based on the physico-chemical parameters which were restricted to measurement of hardness, salinity and determination of potential contamination by sewage. Spatial interpolation (IDW) techniques was used to delineate the locational distribution of groundwater pollutants and through GIS, overlay map of the critical parameters were also produced. Results obtained revealed a very strong spatial dependency of the measured parameters with a calculated (nugget to sill ratio) of less than 25% which is similar with the results of this study.

4. Conclusion
The study was conducted to determine the critical contaminants affecting the overall quality of groundwater around the study area and develop a spatial map using kriging interpolation to visualize the distribution of these contaminants. Results of the study attest to the potential of factor analysis using principal component techniques in the assessment of water quality parameters with a view of selecting the parameters with critical influence on the water quality. More also, geospatial analysis using kriging interpolation revealed that; the water quality parameters showed relatively strong degree of spatial dependency which made it possible to generate the spatial distribution map for the selected water quality parameters.

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