Shape complexity and fractality of fracture surfaces of swelled isotactic polypropylene with supercritical carbon dioxide

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We have investigated the fractal characteristics and shape complexity of the fracture surfaces of swelled isotactic polypropylene Y1600 in supercritical carbon dioxide fluid through the consideration of the statistics of the islands in binary SEM images. The distributions of area $A$, perimeter $L$, and shape complexity $C$ follow power laws $p(A) \sim A^{-(\mu_A+1)}$, $p(L) \sim L^{-(\mu_L+1)}$, and $p(C) \sim C^{-(\nu+1)}$, with the scaling ranges spanning over two decades. The perimeter and shape complexity scale respectively as $L \sim A^{D/2}$ and $C \sim A^q$ in two scaling regions delimited by $A \approx 10^3$. The fractal dimension and shape complexity increase when the temperature decreases. In addition, the relationships among different power-law scaling exponents $\mu_A$, $\mu_B$, $\nu$, $D$, and $q$ have been derived analytically, assuming that $A$, $L$, and $C$ follow power-law distributions.

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I. INTRODUCTION

Polypropylene, being one of the fastest growing engineering plastics, has wide industrial and everyday life applications due to its intrinsic properties such as low density, high melting point, high tensile strength, rigidity, stress crack resistance, abrasion resistance, low creep and a surface which is highly resistance to chemical attack, and so on [1]. Impregnation of nucleating agents allows the polymer to be crystallized at a higher temperature during processing and changes a lot the mechanical and optical performance of the resultant polymer matrix [2,3]. In such nucleating agent impregnating processes, supercritical fluids are well-established swelling solvents whose strength can be tuned continuously from gas-like to liquid-like by manipulating the temperature and pressure. Especially, supercritical carbon dioxide is a good swelling agent for most polymers and will dissolve many small molecules [4,5,6,7], except for some fluoropolymers and silicones [8]. This provides the ability to control the degree of swelling in a polymer [8,9,10,11] as well as the partitioning of small molecule penetrants between a swollen polymer phase and the fluid phase [12,13].

The morphology of fracture surfaces of a material is of great concern and interests in many studies [14,15]. A frequently used tools is fractal geometry pioneered by Mandelbrot’s celebrated work [16]. Specifically, fractal geometry has been widely applied to the topographical description of fracture surfaces of metals [17,18], ceramics [19,20,21], polymers [22,23], concretes [24,25,26,27], alloys [28,29,30,31], rocks [32,33,34], and many others. There are many different approaches adopted in the determination of fractal dimensions of fracture surfaces and the estimated fractal dimensions from different materials vary from case to case and are not universal [35].

Concerning the fractal characterization of the fracture surfaces of polypropylene, it was found that the fractal dimension of the fracture surfaces is less than 2.12 [23]. In this paper, we investigate the shape complexity and fractality of fracture surfaces of nucleating agents impregnated isotactic polypropylene (Y1600) swelled with supercritical carbon dioxide based on the area-perimeter relationship [10,17,32]. We shall see that the fractal dimension of the fracture surface of supercritical dioxide swelled polypropylene is much higher than that without swelling and decreases with increasing temperature.

II. EXPERIMENTAL

The isotactic polypropylene (Y1600) powder we have used has an average diameter of $3 \sim 4 \, mm$. Fifteen grams of isotactic polypropylene powder was melt in an oven at $200 \, ^{\circ}C$ and then made into film with a thickness of 0.3 mm by using a press machine with a pressure of 0.5 GN. The size of the standard film samples was $1 \, mm \times 3 \, mm$ and are refluxed with acetone for twenty four hours to remove the impurities, and then annealed at $200 \, ^{\circ}C$ for two hours. The nucleating agent we used was NA21.

The schematic flow chart of the experimental apparatus is illustrated in Fig. 1. The experimental apparatus consists mainly of a gas cylinder, a gas booster, a digital pressure gauge, an electrical heating bath, and valves and fittings of different kinds. The system was cleaned thoroughly using suitable solvents and dried under vacuum. Isotactic PP films marked from one to twelve were placed in the high pressure cell together with desired amount (1% wt) of nucleating agent (NA21). The system was purged with CO$_2$ and after the system had reached the desired temperature and thermal equilibrium, CO$_2$ was charged until the desired pressure was reached. The im-

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FIG. 1: Schematic flow chart of the experimental apparatus.

pregnination process lasted for four hours and then depressurized the CO\textsubscript{2} from the high pressure cell rapidly. Then the vessel was cooled and opened, and the specimens were taken out for analysis. All of the metallic parts in contact with the studied chemicals were made of stainless steel. The apparatus was tested up to 35 MPa. The total volume of the system is 500 mL.

The experiments were performed at fixed pressure of 7.584 MPa and at four different temperatures of $T = 60\,^\circ\text{C}, 80\,^\circ\text{C}, 100\,^\circ\text{C}, \text{and } 120\,^\circ\text{C}$. For each sample, we took ten SEM pictures at magnification of 5000 or 10000. The pictures have 256 grey levels. Due to the nature of area-perimeter approach, the results are irrelevant to the magnification $[16]$. A grey-level picture was transformed into black-and-white image for a given level set $\epsilon$ according to the criterion that a pixel is black if its grey level is larger than $256\epsilon$ and is white otherwise. The resultant binary image has many black islands. A typical binary image of islands is shown in Fig. 2. In our calculations, we have used eleven level sets from 0.45 to 0.95 spaced by 0.05 for each SEM picture.

FIG. 2: Image of the recognized islands from a typical SEM picture of swelled isotactic polypropylene impregnated with NA21 at level set of 0.7.

We sort the $n$ observations in non-increasing order such that $x_1 \geq x_2 \geq \cdots \geq x_R \geq \cdots \geq x_n$, where $R$ is the rank of the observation. It follows that $nP(x \geq x_R)$ is the expected number of observations larger than or equal to $x_R$, that is,

$$nP(x \geq x_R) = R.$$  \hspace{1cm} (1)

If the probability density of variable $x$ follows a power law that $p(x) \sim x^{-(1+\mu)}$, then the complementary distribution $P(x) \sim x^{-\mu}$. An intuitive relation between $x_R$ and $R$ follows

$$x_R \sim R^{-1/\mu}.$$  \hspace{1cm} (2)

A rigorous expression of (2) by calculating the most probable value of $x_R$ from the probability that the $R$-th value equals to $x_R$ gives $[37]$,

$$x_R \sim \left(\frac{\mu n + 1}{\mu R + 1}\right)^{1/\mu}.$$  \hspace{1cm} (3)

when $\mu R \gg 1$ or equivalently $1 \ll R \leq N$, we retrieve (2). A plot of $\ln x_R$ as a function of $\ln R$ gives a straight line with slope $-1/\mu$ with deviations for the first a few ranks if $x$ is distributed according to a power law of exponent $\mu$.

We note that the rank-ordering statistics is nothing but a simple generalization of Zipf’s law $[16,37,38]$ and has wide applications, such as in linguistics $[39]$, the distribution of large earthquakes $[40]$, time-occurrences of extreme floods $[41]$, to list a few. More generally, rank-ordering statistics can be applied to probability distributions other than power laws, such as exponential or stretched exponential distributions $[42]$, normal or log-normal distributions $[37]$, and so on.

Figure 3 illustrates the log-log plots of the rank-ordered areas $A$ at different temperatures. There is
clear power-law dependence between $A$ and its rank $R$ with the scaling ranges spanning over about two decades. Least-squared linear fitting gives $1/\mu_A = 0.80 \pm 0.05$, $1/\mu_A = 0.63 \pm 0.02$, $1/\mu_A = 0.61 \pm 0.03$, and $1/\mu_A = 0.85 \pm 0.04$ with decreasing temperatures, where the errors are estimated by the r.m.s. of the fit residuals. Using $\sigma_{\mu_A} = \sigma_{1/\mu_A}/\mu_A^2$, we obtain the power-law exponents $\mu_A = 1.24 \pm 0.08$, $\mu_A = 1.59 \pm 0.06$, $\mu_A = 1.64 \pm 0.09$, and $\mu_A = 1.18 \pm 0.05$ with decreasing temperatures.

Figure 4 shows the log-log plots of the rank-ordered perimeters $L$ at different temperatures. There are also clear power-law relations between $L$ and its rank $R$ whose scaling ranges spanning over about three decades. Least-squared linear fitting gives $1/\mu_L = 0.56 \pm 0.02$, $1/\mu_L = 0.51 \pm 0.01$, $1/\mu_L = 0.55 \pm 0.03$, and $1/\mu_L = 0.71 \pm 0.03$ with decreasing temperatures. We thus obtain the power-law exponents $\mu_L = 1.78 \pm 0.07$, $\mu_L = 1.95 \pm 0.05$, $\mu_L = 1.83 \pm 0.12$, and $\mu_L = 1.41 \pm 0.06$ with decreasing temperatures. It is noteworthy that the scaling ranges of the rank-ordering of both $A$ and $P$ are well above two decades broader than most of other experiments with scaling ranges centered around 1.3 orders of magnitude and spanning mainly between 0.5 and 2.0.

### B. Scaling between area and perimeter

If an island is fractal, its area $A$ and perimeter $L$ follow a simple relation,

$$L^{1/D} \sim A^{1/2},$$

where $D$ is the fractal dimension describing the wiggliness of the of the perimeter. For $D = 1$, the perimeter of the island is smooth. For $D = 2$, the perimeter becomes more and more contorted to fill the plane. This area-perimeter relation (4) can also be applied to many islands when they are self-similar in the sense that the ratio of $L^{1/D}$ over $A^{1/2}$ is constant for different islands. It has been used to investigate the geometry of satellite- and radar-determined cloud and rain regions by drawing $A$ against $L$ for different regions in a log-log plot. The slope of the straight line obtained by performing a linear regression of the data gives $2/D$.

Following the same procedure, we plotted for each temperature $L$ against $A$ for all islands identified at different level sets. The resultant scatter plot for a given temperature shows nice collapse to a straight line. The four slopes are $D = 1.34 \pm 0.08$ for $T = 60^\circ C$, $D = 1.32 \pm 0.07$ for $T = 80^\circ C$, $D = 1.31 \pm 0.08$ for $T = 100^\circ C$, and $D = 1.28 \pm 0.09$ for $T = 120^\circ C$, respectively. The errors were estimated by the r.m.s. of the fit residuals.

However, a closer investigation of the scatter plots shows that there are two scaling regions with a kink at around $A = 1000$. To have a better view angle, we adopt an averaging technique for both $A$ and $L$. We insert $n - 2$ points in the interval $[\min(A), \max(A)]$ resulting in $\min(A) = a_0 < a_1 < a_2 < \cdots < a_n = \max(A)$ so that $a_i$’s are logarithmically spaced. We can identify all islands whose areas fall in the interval $[a_i, a_{i+1})$. The geometric means of $L$ and $A$ are calculated for these islands, denoted as $\langle L \rangle$ and $\langle A \rangle$. The calculated means $\langle L \rangle$ and $\langle A \rangle$ are plotted in Fig. 5. The two scaling regions are fitted respectively with two straight lines and we have $D_1 = 1.42 \pm 0.06$ for $T = 60^\circ C$, $D_1 = 1.33 \pm 0.06$ for $T = 80^\circ C$, $D_1 = 1.33 \pm 0.07$ for $T = 100^\circ C$, and $D_1 = 1.30 \pm 0.05$ for $T = 120^\circ C$, in the first region that $\langle A \rangle < 1000$ and $D_2 = 1.96 \pm 0.07$ for $T = 60^\circ C$, $D_2 = 1.89 \pm 0.24$ for $T = 80^\circ C$, $D_2 = 2.15 \pm 0.11$ for
The four estimated values of $D$ are 1.67, 1.79, 1.63, and 1.40 with increasing temperature. The discrepancy from $D_1$ is remarkable (15%, 25%, 18%, and 7%, accordingly). The source of this discrepancy is threefold. Firstly, the power-law distributions of $A$ and $L$ have cutoffs at both ends of small and large values so that the derivation of (3) is not rigorous. Secondly, the determination of the scaling ranges may cause errors. Thirdly, the scaling ranges of the three power laws are not consistent with each other.

C. Shape complexity

To quantify the shape complexity of an irregular fractal object, besides the fractal dimension, there are other relevant measures related to the fractal nature [17, 40]. For a $d$-dimensional hypersphere, its surface and volume are related by

$$S_{d, sph} = \frac{1}{k_d} V_{d}^{(d-1)/d},$$

where

$$k_d = \frac{\Gamma(1/2 + d/2)}{d\pi^{1/2}}.$$ (7)

Since a hypersphere has the maximal enclosed volume among the objects with a given surface $S_d$, we have

$$S_d \geq \frac{1}{k_d} V_{d}^{(d-1)/d}.$$ (8)

Then the following dimensionless ratio

$$C_d \equiv \frac{k_d S_d}{V_{d}^{(d-1)/d}}$$ (9)

describes the irregularity or complexity of the investigated object [48]. It follows that $1 \leq C_d \leq \infty$. The limits are reached at $C_d = 1$ for hyperspheres and $C_d = \infty$ for fractals. For real fractal objects, sometimes known as prefractals [17], the scaling range is not infinite. The shape complexity $C_d$ is thus finite. In the case of $d = 2$, we have

$$1 \leq C \leq \frac{L}{2 \pi^{1/2} A^{1/2}} \leq \infty.$$ (10)

This dimensionless measure of shape complexity has been applied to scalar field of concentration in turbulent jet at high Reynolds numbers [48, 49].

In Fig. 5 is shown the log-log plot of the rank-ordered shape complexity $C$ at different temperatures. The power-law dependence $C \sim R^{1/\nu}$ implies a power-law probability density of shape complexity, that is,

$$p(C) \sim C^{-(\nu+1)}.$$ (11)

We find that $\nu = 3.97 \pm 0.11$ for $T = 60{\circ}C$, $\nu = 4.61 \pm 0.17$ for $T = 80{\circ}C$, $\nu = 4.93 \pm 0.20$ for $T = 100{\circ}C$, and $\nu = 5.22 \pm 0.17$ for $T = 120{\circ}C$. The mean logarithmic complexity $\langle \ln C \rangle = 1/(\nu-1)$ [48] is calculated to be 0.34, 0.28, 0.25, and 0.24 with increasing temperature. In other words, the shape complexity of the islands decreases with increasing temperature. This is consistent...
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consideration of the statistics of the islands in binary images. At a given temperature, the distributions of area
and perimeter of the islands are found to follow power laws spanning over two decades of magnitude, via rank-ordering statistics. The well-established power-law scaling between area and perimeter shows the overall self-similarity among islands when \( A < 10^3 \). This shows that the fracture surface is self-similar. For large islands, the fractal dimension is estimated to be close to 2 which can be explained by the fact that most of the large islands are strip-shaped. The fracture surface is rougher at low temperature with larger fractal dimension.

We have also investigated the shape complexity of the fracture surfaces using a dimensionless measure \( C \). The distribution of shape complexity is also found to follow a power law spanning over two decades of magnitude. The shape complexity increases when the island is larger. There are two power-law scaling ranges between \( C \) and \( A \) delimited around \( A = 10^3 \), corresponding to the two-regime area-perimeter relationship. The exponent \( q_1 \) serves as an inverse measure of overall shape complexity, which is observed to increases with temperature. This is consistent with the change of fractal dimension at different temperatures.

Furthermore, the relationships among different power-law scaling exponents of the probability distributions \( (\mu_A, \mu_B, \nu) \), of area-perimeter relation \( (D) \), and of complexity-area relation \( (q) \) have been derived analytically. However, these relations hold only when the probability distributions of \( A, L, \) and \( C \) follow exactly power laws. In the present case, we observed remarkable discrepancy between numerical and analytical results.

IV. CONCLUSION

We have investigated the fractal characteristics of the fracture surfaces of swelled isotactic polypropylene Y1600 impregnated with nucleating agent NA21 through consideration of the statistics of the islands in binary images. At a given temperature, the distributions of area

FIG. 6: Log-log plot of the rank-ordered shape complexity \( C \) at different temperatures. The plots are translated vertically for clarity. The solid lines are linear fits to the data at \( 40 \leq R \leq 3890 \) for \( T = 120 \, ^\circ C \), \( 24 \leq R \leq 5000 \) for \( T = 100 \, ^\circ C \), \( 40 \leq R \leq 3000 \) for \( T = 80 \, ^\circ C \), and \( 25 \leq R \leq 1500 \) for \( T = 60 \, ^\circ C \), respectively.

FIG. 7: Plots of \( \langle C \rangle \) with respect to \( \langle A \rangle \) at different temperatures. The plots are translated vertically for clarity. We note that \( \langle C \rangle = 2/\pi^{1/2} \) when \( \langle A \rangle = 1 \) in all the four plots.

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[1] E. P. Moore, ed., *Polypropylene Handbook* (Hanser Gardner, New York, 1996).
[2] T. Xu, H. Lei, and C.-S. Xie, Materials and Design 24, 227 (2003).
[3] J. O. Iroh and J. P. Berry, Eur. Polymer J. 32, 1425 (1996).
[4] J. S. Chiou, J. W. Barlow, and D. R. Paul, J. Appl. Polym. Sci. 30, 2633 (1985).
[5] Y. Kamiya, T. Hirose, Y. Naito, and K. Mizoguchi, J. Polym. Sci. B 26, 159 (1988).
[6] A. R. Berens, G. S. Huvard, R. W. Korsmeyer, and F. W. Kunig, J. Appl. Polym. Sci. 46, 231 (1992).
[7] Y. Kamiya, K. Mizoguchi, K. Terada, Y. Fujiwara, and J. S. Wang, Macromolecules 31, 472 (1998).
[8] W. H. Tuminello, G. T. Dee, and M. A. McHugh, Macromolecules 28, 1506 (1995).
[9] G. K. Fleming and W. J. Koros, Macromolecules 19, 2285 (1986).
[10] D. S. Pope and W. J. Koros, J. Polym. Sci. B 34, 1861 (1996).
[11] P. D. Condo, S. R. Sumpter, M. L. Lee, and K. P. Johnston, Ind. Eng. Chem. Res. 35, 1115 (1996).
[12] M. F. Vincent, S. G. Kazarian, and C. A. Eckert, AIChE J. 43, 1838 (1997).
[13] S. G. Kazarian, M. F. Vincent, B. L. West, and C. A. Eckert, J. Supercrit. Fluids 13, 107 (1998).
[14] J. Fineberg and M. Marder, Phys. Rep. 313, 1 (1999).
[15] G. P. Cherepanov, A. S. Balankin, and V. S. Ivanova, Engin. Fract. Mech. 51, 997 (1995).
[16] B. B. Mandelbrot, *The Fractal Geometry of Nature* (W. H. Freeman and Company, New York, 1983).
[17] B. Mandelbrot, D. Passoja, and A. Paullay, Nature 308, 721 (1984).
[18] U. Wendt, K. Steibe-Lange, and M. Smid, J. Microscopy 207, 169 (2002).
[19] J. J. Mecholsky, D. E. Passoja, and K. S. Feinberg-Ringel, J. Am. Ceram. Soc. 72, 60 (1989).
[20] J. Y. Thompson, K. J. Anusavice, B. Balasubramaniam, and J. J. Mecholsky, J. Am. Ceram. Soc. 78, 3045 (1995).
[21] A. Celli, A. Tucci, L. Esposito, and P. Carlo, J. Euro. Ceram. Soc. 23, 469 (2003).
[22] C. T. Chen and J. Runt, Polym. Commun. 30, 334 (1989).
[23] J. Yu, T. Xu, Y. Tian, X. Chen, and Z. Luo, Materials and Design 23, 89 (2002).
[24] M. A. Issa, M. A. Issa, M. S. Islam, and A. Chudnovsky, Engineering Fracture Mechanics 70, 125 (2003).
[25] A. Yan, K.-R. Wu, D. Zhang, and W. Yao, Cem. Concr. Comp. 25, 153 (2003).
[26] L. Dougan and P. Addison, Com. Concr. Res. 31, 1043 (2001).
[27] Y. Wang and S. Diamond, Com. Concr. Res. 31, 1385 (2001).
[28] V. I. Betekhtin, P. N. Butenko, V. L. Gilyarov, V. E. Korsukov, A. S. Luk’yanenko, B. A. Obidov, and V. E. Khartsiev, Tech. Phys. Lett. 28, 26 (2002).
[29] C. Shek, G. Lin, K. Lee, and J. Lai, J. Non-Crystalline Solids 224, 244 (1998).
[30] X. Wang, H. Zhou, Z. Wang, M. Tian, Y. Liu, and Q. Kong, Materials Sci. Engin. A 266, 250 (1999).
[31] A. Eftekhari, Appl. Surface Sci. 220, 343 (2003).
[32] T. Babadagli and K. Develi, Theore. Appl. Fracture Mech. 39, 73 (2003).
[33] H. Xie, H. Sun, Y. Ju, and Z. Feng, Int. J. Solids Struct. 38, 5765 (2001).
[34] H. W. Zhou and H. Xie, Surface Rev. Lett. 10, 751 (2003).
[35] S. Wang, Physica B 348, 183 (2004).
[36] S. Lovejoy, Science 216, 185 (1982).
[37] D. Sornette, *Critical Phenomena in Natural Sciences - Chaos, Fractals, Self-organization and Disorder: Concepts and Tools* (Springer, Berlin, 2000), 1st ed.
[38] G. K. Zipf, *Human Behavior and The Principle of Least Effort* (Addison-Wesley Press, Massachusetts, 1949).
[39] B. B. Mandelbrot, Word 10, 1 (1954).
[40] D. Sornette, L. Knopoff, Y. Kagan, and C. Vanneste, J. Geophys. Res. 101, 13883 (1996).
[41] A. Mazzarella and F. Rapetti, J. Hydrology 288, 264 (2004).
[42] J. Laherrere and D. Sornette, Eur. Phys. J. B 2, 525 (1998).
[43] O. Malcai, D. Lidar, O. Biham, and D. Avnir, Phys. Rev. E 56, 2817 (1997).
[44] D. Avnir, O. Biham, D. Lidar, and O. Malcai, Science 279, 39 (1998).
[45] H. J. Catrakis and P. E. Dimotakis, Phys. Rev. Lett. 80, 968 (1998).
[46] J. Kondev, C. L. Henley, and D. G. Salinas, Phys. Rev. E 61, 104 (2000).
[47] P. S. Addison, Fractals 8, 147 (2000).
[48] H. J. Catrakis, R. C. Aguirre, J. Ruiz-Plancarte, and R. D. Thayne, Phys. Fluids 14, 3891 (2002).