Impact of N-Terminal Acetylation of α-Synuclein on Its Random Coil and Lipid Binding Properties

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3 Supporting Information

ABSTRACT: N-Terminal acetylation of α-synuclein (αS), a protein implicated in the etiology of Parkinson’s disease, is common in mammals. The impact of this modification on the protein’s structure and dynamics in free solution and on its membrane binding properties has been evaluated by high-resolution nuclear magnetic resonance and circular dichroism (CD) spectroscopy. While no tetrameric form of acetylated αS could be isolated, N-terminal acetylation resulted in chemical shift perturbations of the first 12 residues of the protein that progressively decreased with the distance from the N-terminus. The directions of the chemical shift changes and small changes in backbone $\gamma_{H,H}$ couplings are consistent with an increase in the \( \alpha \)-helicity of the first six residues of αS, although a high degree of dynamic conformational disorder remains and the helical structure is sampled <20% of the time. Chemical shift and $^J_{HH}$ data for the intact protein are virtually indistinguishable from those recorded for the corresponding N-terminally acetylated and nonacetylated 15-residue synthetic peptides. An increase in \( \alpha \)-helicity at the N-terminus of αS is supported by CD data on the acetylated peptide and by weak medium-range nuclear Overhauser effect contacts indicative of \( \alpha \)-helical character. The remainder of the protein has chemical shift values that are very close to random coil values and indistinguishable between the two forms of the protein. No significant differences in the fibrillation kinetics were observed between acetylated and nonacetylated αS. However, the lipid binding properties of αS are strongly impacted by acetylation and exhibit distinct behavior for the first 12 residues, indicative of an initiation role for the N-terminal residues in an “initiation—elongation” process of binding to the membrane.

The link between mutations in the gene encoding α-synuclein (αS) and familial early onset Parkinson’s disease1,2 has stimulated a very wide array of biophysical studies of this protein.3,4 In aqueous solution, this 140-residue protein adopts a dynamically disordered backbone conformation, but its \(^{13}\text{C}\)\(^{13}\text{C}\) chemical shifts also have been interpreted as being indicative of a slight propensity to adopt \( \alpha \)-helical torsion angles for several sections of its backbone.5,6 Electrostatic interaction between the, on average, 100 positively charged N-terminal residues and the negatively charged C-terminal tail results in a hydrodynamic radius \( R_h \approx 28\) Å\(^7\) that is smaller than expected for a true random coil (\( \sim 37\) Å) but larger than the value of \( \sim 15\) Å for a globular protein with the mass of αS.

A recent study suggests that when expressed in mammalian cells, αS can also form a stable helical tetramer,8 and the same conclusion was reached for a form of αS that included an N-terminal decapetide extension, remaining after cleavage of its GST tag.9 In both cases, αS was purified without the heating step, usually employed in αS purification as an effective way to precipitate the vast majority of other, folded proteins. However, the conclusions of both studies remain the subject of much debate.10

The N-terminal 100-residue segment of αS has a high affinity for negatively charged lipids11,12 and adopts an \( \alpha \)-helical conformation when bound to the surface of unilamellar vesicles (SUVs) or detergent micelles.13–17 However, virtually all biophysical studies of αS conducted to date have focused on bacterially overexpressed protein, which lacks post-translational acetylation of the N-terminal residue, a modification commonly found in mammalian proteins.18 Indeed, extensive N-terminal acetylation of αS was detected when it was isolated from erythrocytes.9 Considering the critical role of the N-terminal αS residues in its interaction with phospholipids,18 it therefore is of interest to evaluate the impact of N-terminal acetylation. A very recent study reports that N-terminal acetylation is responsible for the formation of folded, \( \alpha \)-helical αS tetramers.21

Selective chemical acetylation of bacterially expressed αS is challenging because of the many Lys amino groups in the protein. However, a recently developed recombinant expression system22 includes a plasmid for overexpression of the requisite acetylation enzyme NatB and permits bacterial expression of N-terminally acetylated and isotopically enriched αS, needed for NMR studies. Using the recently described “soft” purification protocol,22 we were unable to generate αS samples of sufficient purity for detailed biophysical studies. In an unsuccessful
attempt to detect the putative tetramer, we also employed an aS construct extended with a C-terminal His tag, allowing for effective protein purification by using a nickel affinity column, thereby eliminating the heat denaturation step.

Here, we report primarily on the impact of N-terminal acetylation on the structural and lipid binding properties of wild-type aS (lacking the His tag), as viewed by NMR and CD spectroscopy. We evaluate the effect of the acetylation both on chemical shifts, which are extremely sensitive to even minute structural differences, and on $J_{\text{HN-1H}}$ couplings that are accurate residue-specific reporters of the distribution of the backbone torsion angle $\phi$. Our results show that in the absence of lipids the effect of acetylation is completely restricted to the 12 N-terminal residues. We therefore also evaluate the impact of acetylation on a 15-residue N-terminal peptide fragment of aS by both CD and NMR spectroscopy. Although the structural impact of acetylation remains strictly limited to transient formation of a very short $\alpha$-helical segment at the very N-terminus of the protein, we find a strong increase in lipid binding affinity for acetylated aS.

## MATERIALS AND METHODS

### Protein Expression and Purification

Expression and purification of nonacetylated wild-type (WT) aS were conducted largely as described previously. Briefly, bacteria were grown in M9 medium at 37 °C to an OD$_{600}$ of 0.6, and the protein production was induced by addition of 1 mM IPTG. Cells were harvested by centrifugation 3 h after induction. The cell pellet was immediately resuspended in 20 mL of BOG buffer matching that described by Trexler and Rhoades [100 mM HEPES, 20 mM NaCl, 10% glycerol, and 0.1% BOG (pH 7.4)] with the addition of one full Complete protease inhibitor tablet (Roche). The cell suspension was lysed using a pressure cell, and the lysate then was cleared by centrifugation at 50000g for 30 min. The supernatant was loaded on a 5 mL Ni column. The column was washed with BOG buffer, including 20 mM imidazole. Target protein was then eluted with BOG buffer and 250 mM imidazole, yielding 10 mL of eluate. This solution was dialyzed overnight at 4 °C into a buffer better suited for NMR measurements ([1× PBS, 5% glycerol, and 0.05% BOG (pH 7.4)]. On the basis of the UV absorption measurement, the final protein concentration was 150 μM and the total protein yield was 22 mg. NMR measurements were performed on the dialyzed protein after the addition of 7% D$_2$O and a decrease in the pH to 6 by titration with a 0.1 M HCl stock solution. The time between lowering the pH and the start of the NMR experiments was ~15 min.

### Preparation of SUVs

Phospholipids were purchased from Avanti Polar Lipids (Alabaster, AL) as lyophilized powders of pure DOPC and a DOPC/DOPC mixture with a 5:3:2 weight ratio (coagulation reagent 1). A lipid mixture that consisted of approximately 15% DOPS, 25% DOPE, and 60% DOPC was also prepared by codissolving equal amounts of coagulation reagent 1 and DOPC in chloroform followed by solvent evaporation under a stream of N$_2$ gas. SUVs were prepared as described previously in 20 mM sodium phosphate buffer (pH 6, 10% [w/v]).

### NMR Spectroscopy

Diffusion measurements were performed on a 600 MHz Bruker spectrometer equipped with a triple-gradient room-temperature probe using the water-sLED experiment with observation optimized for the methyl region of the spectrum. The samples contained 100 μM protein and 5 mM dioxane (internal reference) in 20 mM sodium phosphate buffer (pH 6). Experiments were performed at 288 K using a diffusion delay of 300 ms and 1 ms $x-y$-gradient pulses with intensities set to 2, 8, 10, 12, 20, 25, 30, 35, 40, 44, 46, and 48% of the maximal value. An acquisition time of 200 ms was used, and 2048 scans with a 1.6 s interscan delay were collected for each gradient strength. Peak attenuations were measured in the methyl region of the spectrum, by scaling relative to the spectrum collected for the weakest gradient, using the dual-mode feature of Bruker XwinNMR, and hydrodynamic radii were extracted using the dioxane hydrodynamic radius (2.12 Å) as an internal reference.

For the backbone chemical shift assignments, three-dimensional (3D) TROSY-HNCO and TROSY-HNCAcb spectra were recorded on 15N- and 13C-labeled WT aS (0.35 mM nonacetylated aS and 0.45 mM N-terminally acetylated aS) at 288 K, using a 600 MHz Bruker Avance II spectrometer.
running Topspin 2.1 and equipped with a z-axis gradient TCI cryogenic probe. All data were collected with two scans per free induction decay. The HNCO spectra comprised 70° × 200° × 575° complex points, for acquisition times of 70.7, 150, and 76.7 ms in the 15N, 13C, and directly detected 1H dimensions, respectively. An acquisition time in the 13C dimension much longer than the 1JCC refocusing INEPT delay, typically around 30 ms, was achieved by using the mixed-time (MT) evolution approach without requiring additional pulses. With an interscan delay of 0.8 s, the total measuring time for each HNCO spectrum was approximately 1.5 days.

The TROSY-HNCA spectra were collected using the acquisition parameters listed above in the 15N and 1H dimensions. The 13C chemical shift was recorded using a 28° 1H frequency of 900 MHz. In total, nine constant-time delays with durations of 50, 60, 75, 90, 140, 180, 210, 240, and 280 ms were used, which samples a J modulation curve with up to two zero crossing points for the majority of residues. Depending on the length of the JHH modulation time, the data matrix size ranges from 79° × 1024° in the 15N and 1H dimensions for the shortest constant-time duration to 579° × 1024° points for the longest constant-time duration, with a 103.2 ms 1H acquisition time in all the experiments. Cross-peaks were picked from the spectrum with the longest 15N acquisition time and therefore the highest resolution, unless it was near the zero crossing point and the peaks were too weak for a reliable determination of the peak positions, in which case the next best resolved spectrum was used. With the peak positions accurately determined, the autoFit.tcl routine in the NMRPipe software package was applied to extract the peak intensity in each spectrum, keeping the position fixed. Nonlinear least-squares fitting of the J-modulated peak intensities to eq 4 of ref 28 was performed to obtain the 3JHH coupling.

Two-dimensional (2D) TOCSY and NOESY spectra were recorded at 278 K on peptide samples containing 1.6 mM nonacetylated and 1.0 mM acetylated N-terminal 15-residue peptides, both amidated at V15, using a 900 MHz Bruker Avance III spectrometer with a cryogenic probe. For the NOESY spectra, a mixing time of 300 ms was used. In all the spectra, 512° × 2048° complex points in the indirect and direct dimensions were collected for acquisition times of 69.6 and 227 ms. The interscan delay was set to 1.5 s, with four and eight scans per free induction decay, yielding total measuring times of 2 and 4.5 h for the 2D TOCSY and NOESY spectra, respectively.

TROSY-HSQC spectra were recorded at 1H frequencies of 500 and 800 MHz on Bruker spectrometers equipped with cryogenic probes. At 500 MHz, 200° × 560° complex points were collected for acquisition times of 165 (τ1) and 80 ms (τ2). At 800 MHz, 300° × 900° complex points were collected for acquisition times of 154 (τ1) and 80 ms (τ2).

**CD Measurements.** All CD measurements were taken at 20 °C, using a 1 mm path-length cuvette. Samples contained 20 mM sodium phosphate (pH 6), 150 mM NaCl buffer, and either 100 μM 15-residue peptides or 10 μM full-length protein.

**Fibrillation Experiments.** Experiments were performed using a Tectan Infinite 200 Pro multimode microplate reader. The sample volume was 200 μL per well. Sample conditions were as follows: 200 μM WT aS, either acetylated or nonacetylated, PBS buffer, pH 7.4. The plate was shaken at 88 rpm and 37 °C. A 1 mm diameter glass bead was placed into each sample well to accelerate fibrillation. Fibril formation was followed using the fluorescence of thioflavin T (ThT), which was added at a concentration of 10 μM to all samples. The instrument took fluorescence readings of all samples at regular intervals of 50 min for a total duration of 92 h. The excitation wavelength was set to 415 nm, and the ThT emission was measured at 480 nm. Four samples for acetylated and three for nonacetylated aS were measured.

### RESULTS AND DISCUSSION

The recently introduced bacterial expression system coexpresses the fission yeast NatB enzyme complex together with the protein of interest. It provides an effective method for obtaining N-terminal acetylation of proteins such as aS, which starts with a Met-Asp pair at its N-terminus. Using the regular bacterial strain BL21(DE3), we obtained yields of acetylated WT aS that were comparable to amounts normally harvested for nonacetylated protein (∼20 mg/L) when expressed in minimal media, required for incorporation of 15C and 15N stable isotopes. With no resonances visible in the 15N-1H HSQC NMR spectrum at the positions of the nonacetylated protein, levels of acetylation were established to be >98%. However, when an attempt was made to express perdeuterated aS, which can generate extremely well resolved 15N-1H correlation spectra for aS, the level of acetylation was at most 40% (for more details, see Materials and Methods).

All results presented in this study therefore are for fully protonated material.

**Effect of Acetylation on NMR Characteristics of aS.** A comparison of the 15N-1H HSQC NMR spectra of N-terminally acetylated and nonacetylated WT aS reveals the impact of acetylation to be strictly limited to the first 12 residues of the protein, and both line widths and resonance positions are essentially indistinguishable for the remainder of the protein, despite the very high precision at which these values were recorded (Figure 1). Backbone assignments (Tables S1 and S2 of the Supporting Information) were obtained from nonuniformly sampled HNCACB 3D spectra. Evaluation of the actual chemical shift values in terms of secondary structure propensity, using the latest sets of sequence, pH, and temperature correction factors, shows root-mean-square differences (rmsds) from random coil values that are even considerably smaller (Figure S2 of the Supporting Information) than previously reported for both aS and other intrinsically unstructured proteins.

The newer chemical shift evaluation method of Camilloni et al. identifies secondary structure by using all available backbone chemical shifts as input parameters. With this program, δ2D, the population of α-helix...
for αS in free aqueous solution is estimated to be even lower than initially deduced from older, less advanced approaches (Figure 2). This result is consistent with the random coil characteristics of its CD spectrum recorded in the absence of lipids (Figure 3A). However, δ2D indicates a small but distinct increase in the population of α-helix near the N-terminus for the acetylated form of the protein (Figure 2). Although the effect of N-terminal acetylation on the chemical shifts of residues 6–12 rapidly decreases with the distance from the N-terminus (Figure 1), it is rather remarkable that the effect of this small covalent modification in a disordered protein propagates as far as residue 12, an effect explained by the cooperative formation of a transient short α-helix of the N-terminal residues.

Remarkably, there is no discernible impact of the acetylation on the chemical shifts of the acidic C-terminal tail of the protein, despite the decrease in positive charge and increase in helical population at the N-terminus. An interaction between the net positively charged N-terminus and the acidic C-terminal region of the protein was previously established on the basis of paramagnetic relaxation experiments and was postulated to be primarily electrostatic in nature. The presence of such a long-range interaction is also reflected in a hydrodynamic volume

Figure 1. Impact of N-terminal acetylation on the NMR spectrum of αS. (A) Overlay of a small expanded region of the 800 MHz HSQC spectra of nonacetylated (black) and acetylated (red) αS. The pronounced change in the peak position for Ser9 is marked with an arrow. For the overlay of full spectra, see Figure S1 of the Supporting Information. Differences in chemical shifts in parts per million between acetylated and nonacetylated αS for different backbone atoms as a function of residue number: (B) 13Cα, (C) 13Cβ, (D) 13C′, (E) 15N, and (F) 1H N. Pairwise rmsds calculated over residues 13–140 are 0.008 ppm (13Cα), 0.007 ppm (13Cβ), 0.005 ppm (13C′), 0.019 ppm (15N), and 0.002 ppm (1H N).

Figure 2. Population of α-helix for nonacetylated (black) and acetylated (red) WT αS as derived from the measured 13Cα, 13Cβ, 13C′, 15N, and 1H N chemical shifts using δ2D.31

Figure 3. CD data for the interaction of nonacetylated (black) and acetylated (red) WT αS with lipid vesicles. (A) Solid lines represent CD spectra obtained at a 75:1 lipid:protein molar ratio. Dashed lines show CD spectra in the absence of lipids. (B) Graph showing the change in CD signature at 222 nm, reflecting the amount of α-helical structure, as a function of lipid:protein molar ratio. Buffer conditions were 20 mM phosphate (pH 6) and 150 mM NaCl. Measurements were performed on samples containing 10 μM protein. The lipid consisted of 30% DOPS, 50% DOPE, and 20% DOPC.
that is ∼2.3 times smaller than expected for a true random coil of 140 residues.7,32 The absence of any detectable chemical shift perturbation for residues 13–140 upon losing the positive N-terminal charge indicates that the acetylation has no discernible effect on the time-averaged backbone angles of these residues, highlighting the absence of specific interactions between the C-terminal region and the N-terminus. However, very weak, nonspecific interactions between the N-terminal and C-terminal regions are reflected in a slight decrease in the global electrostatic compaction of the protein, as indicated by a small ∼1.5% decrease in the rate of translational diffusion upon acetylation of WT aS (Figure 4). The 1.5% decrease in translational diffusion corresponds to a 1.5% (0.5 Å) increase in the hydrodynamic radius and requires only extremely small (∼1°) changes in the time-averaged backbone torsion angle distributions. Therefore, the absence of any significant difference in backbone torsion angles between the acetylated and nonacetylated forms of the protein, deduced from the indistinguishable chemical shifts, is not inconsistent with the slight difference in the hydrodynamic radii of the two forms of the protein.

Next to chemical shifts, $J_{HN-Hα}$ couplings are particularly reliable quantitative reporters of the time-averaged distribution of the backbone torsion angles, $ϕ$.33 For folded proteins with known structures, the quantitative relation between these torsion angles and $J_{HN-Hα}$ couplings is defined by the empirical Karplus equation. Recent work has demonstrated that this equation is remarkably robust and typically is limited by the accuracy of the atomic coordinates that define the $ϕ$ angles and by small out-of-peptide-plane deviations of the amide protons.34 When atomic coordinates were used that had been refined by the use of residual dipolar couplings, the level of agreement between $J_{HN-Hα}$ couplings and backbone torsion angles decreased to within the 0.35 Hz uncertainty of the measurement, indicating that no correction for amino acid type (excluding Gly) or H-bonding is needed.34 We therefore measured a nearly complete set of $J_{HN-Hα}$ couplings in both acetylated and nonacetylated aS. With the exception of small decreases in $J_{HN-Hα}$ for the first few residues in the protein upon acetylation (Table S4 of the Supporting Information), the values measured for the two forms of the protein are essentially indistinguishable (rmsd of 0.05 Hz).

It is interesting to note, however, that the different residue types show distinct clustering of their $J_{HN-Hα}$ values (Table S4 of the Supporting Information), with the smallest values found for Ala residues ($J_{HN-Hα} = 5.64$ ± 0.26 Hz, excluding C-terminal A140) and the largest values found for β-branched residues such as Val ($J_{HN-Hα} = 7.60$ ± 0.29 Hz), reflecting the natural tendencies of these amino acids to populate more helical or extended regions of Ramachandran space in intrinsically unstructured peptides and proteins. The first six residues of acetylated aS show decreases relative to these random coil values that fall outside the standard deviation, again indicative of $α$-helix. Residues following G7 do not show such an effect.

**Effect of Acetylation on an N-Terminal 15-Residue Peptide Fragment.** As the impact of acetylation appears to be restricted to the first few residues of aS, we also studied the synthetic 15-residue N-terminal fragment of the protein in the absence and presence of N-terminal acetylation. Because of their much smaller number of residues, these peptides exhibit far less $^1$H resonance overlap than the full-length protein and make it possible to study their structure by the conventional 2D NOEY method. With the exception of residues 13–15, proximate to the C-terminus of the peptide, the chemical shifts of the two peptides match those seen in the corresponding full-length proteins very closely, lending validity to the study of these peptides. The nonacetylated peptide shows essentially random coil characteristics. Its 2D NOEY spectrum lacks medium-range ($i, i + n; 2 \leq n \leq 5$) NOEs at the lowest contour level. By contrast, weak $d_{αβ}(i,i+3)$, $d_{αβ}(i,i+3)$, and $d_{αβ}(i,i+4)$ NOEs point to $α$-helical character for the first six residues of the acetylated peptide (Figure S3 of the Supporting Information). However, upon quantitative comparison of the intraresidue $d_{αβ}(i,i)$ and sequential $d_{αβ}(i−1,i)$ NOE intensities, these ratios remain considerably below unity. As pointed out previously,35,36 the $d_{αβ}(i,i)/d_{αβ}(i−1,i)$ NOE intensity ratio is very sensitive to the value of the $ψ$ angle of residue $i − 1$ and ranges from ∼6 for $α$-helix to ∼0.25 for β-sheet. Ratios observed in the absence of acetylation are very close to values of ∼0.35, typical of fully disordered proteins. A decrease of ∼50% in the intensity of the sequential $d_{αβ}(i−1,i)$ NOEs for residues V3 and F4 increases the NOE intensity ratio to >0.5 (Figure 5), consistent with a shift to an increased population of helical backbone angles compared to random coil. However,
the fact that \(d_{\text{ON}}(i)/d_{\text{ON}}(i-1)\) ratios remain much below unity clearly indicates that the population of \(\alpha\)-helical conformations remains low, fully consistent with the analysis of chemical shifts, described above.

Only a very slight increase in \(\alpha\)-helicity is observed in the CD spectrum of the full aS protein upon acetylation (Figure 3A), because the CD signal remains dominated by the random coil character of the \(\sim 95\%\) of the chain that is not impacted by the N-terminal acetylation. However, upon comparison of the CD spectra of the N-terminal 15-residue peptides, an increase in \(\alpha\)-helical character is clearly observed upon acetylation (Figure S4 of the Supporting Information). Consistent with the NMR results, the degree of \(\alpha\)-helicity remains low, however.

**Estimated Increase in the Helicity of the First Five Residues.** N-Terminal acetylation is known to stabilize \(\alpha\)-helices.\(^{37,38}\) For the first few residues of N-terminally acetylated aS in the absence of lipids and detergents, comparison of \(\delta_{\text{HN-Ht}}\) couplings as well as \(^{13}\text{C}\) and \(^{13}\text{C}^\circ\) chemical shifts with those of the nonacetylated protein shows a small increase in the propensity for \(\alpha\)-helical structure. Acetylation results in an \(\sim 0.5\) ppm increase in the \(^{13}\text{C}^\circ\) chemical shifts for the first five residues and progressively smaller increases for the following few residues. Assuming an \(\sim 3\) ppm secondary chemical shift for an ideal \(\alpha\)-helix, this indicates an \(\sim 17\%\) increase in the \(\alpha\)-helical population for the first five residues.

Upon acetylation, \(\delta_{\text{HN-Ht}}\) decreases by \(\sim 0.5\) Hz for residues 3–5, whereas the value for K6 (6.1 Hz) falls about two standard deviations or 0.5 Hz below the values observed for Lys residues in the remainder of the chain for both forms of the protein (Table S4 of the Supporting Information). With an average \(\delta_{\text{HN-Ht}}\) of 4 Hz expected for an ideal \(\alpha\)-helix and a \(\delta_{\text{HN-Ht}}\) of \(\sim 7\) Hz for random coil, the approximate increase in the level of \(\alpha\)-helical conformation is also estimated to be \(\sim 17\%\).

CD measurements show a markedly stronger helical signature for the acetylated N-terminal 15-residue peptide than for the nonacetylated peptide (Figure S4 of the Supporting Information). If this increase is attributed solely to a change in helical content, then the average increase in helicity is \(\sim 6\%\) (according to JFit). Considering that on the basis of our NMR chemical shift perturbation this change can be attributed to the first five residues, the \(6\%\) increase in average helicity corresponds to an \(\sim 18\%\) increase in helicity for these five residues.

Thus, the three independent indicators of helical propensity all indicate an increase in helicity for the first five residues of \(\sim 17\%\), a number that is slightly larger than the smoothed \(11\%\) increase reported by the \(\delta2D\) program of Camilloni et al.\(^{31}\) which takes into account all backbone secondary chemical shifts.

**Impact of Acetylation on Lipid Binding.** Interactions between aS and a variety of different types of lipid vesicles have previously been studied by optical spectroscopic methods, in particular CD,\(^{11,13}\) as well as by both continuous wave and pulsed EPR,\(^{14,15,39–42}\) and solution NMR,\(^{5,29}\) spectroscopies. All of these techniques indicate a transition from a disordered to an \(\alpha\)-helical state upon lipid binding, with some debate remaining about whether the first \(\sim 100\) residues form a single contiguous helix or whether aS adopts a flexibly tethered two-helix structure, as was seen in the presence of small SDS micelles.\(^{16,43}\) The various studies unequivocally agreed on the important role of negative membrane charge in the interaction, with the highest affinity for lipids with phosphatidylserine headgroups and a very low affinity for neutral, zwitterionic phosphatidylcholine lipids, an observation solidified by fluorescence correlation spectroscopy.\(^{12}\)

Even when using very dilute suspensions of negatively charged small unilamellar vesicles (SUAVs), corresponding to a lipid:protein molar ratio as low as unity, significant attenuation of NMR resonance intensity of the N-terminal residues of aS was observed.\(^{17}\) Remarkably, the attenuated N-terminal HSQC cross-peaks are not shifted relative to those seen in lipid-free samples and show an only very small degree of line broadening, indicating a slow exchange process between the free, disordered state and the lipid-bound state. The absence of attenuation seen for the C-terminal residues indicates that these residues retain their dynamically disordered, random coil character even when the corresponding N-terminal amides have converted to an NMR-invisible “dark state”, induced by lipid binding.\(^{17,29}\)

Here, we evaluated the impact of N-terminal acetylation of aS on lipid binding by both CD and NMR spectroscopy. CD measurements were taken under conditions of 20 mM sodium phosphate buffer (pH 6) and 150 mM NaCl at 20 °C, using SUAVs composed of 30% DOPS, 50% DOPE, and 20% DOPC. This commercially available lipid mixture is convenient for generating highly stable and reproducible SUAVs and includes the headgroups of some of the most abundant phospholipids found in synaptic vesicles.\(^{44}\) At an intermediate lipid:protein molar ratio of 75, the CD spectra show a substantially higher \(\alpha\)-helical content for acetylated aS (Figure 3A). A full titration curve, probing the effect of lipid binding on helicity over a wide range of lipid:protein stoichiometries, indicates that acetylated aS has a roughly 2-fold higher affinity for the negatively charged lipid vesicles than does nonacetylated protein (Figure 3B).

NMR experiments were performed under the same buffer conditions but at 15 °C, with lipid vesicles composed of 15% DOPS, 25% DOPE, and 60% DOPC. Vesicles with lower charges were used because this decreases their affinity for aS and thereby weakens the effect of crowding on their surface when aS binds to them. These lower-affinity vesicles are less suitable for CD measurements because of interference from SUV scattering at high lipid concentrations but are well suited for NMR measurements. As one can see in Figure 6, the general shape of the attenuation profile observed for the nonacetylated protein shows the same stepwise attenuation profile as...
previously reported by Bodner et al. for the vesicles carrying a higher charge density. Remarkably, however, the acetylated aS shows a distinct additional attenuation for the first 12 residues, followed by the transition to the characteristic flat profile for residues 25–90 (Figure 6). The same average attenuation of residues 25–90 is achieved for acetylated aS at 2-fold lower lipid concentration, compared to that of the nonacetylated protein. This observation is consistent with the ~2-fold higher lipid affinity of acetylated aS, indicated by the CD measurements.

Effect of Acetylation on aS Fibrillation. The impact of N-terminal acetylation on the fibril forming propensity of aS, in the absence of lipids, was investigated by measurement of the fibrillation kinetics of the two forms of the protein, using a Tecan Infinite 200 Pro multimode microplate reader to monitor ThT fluorescence. Readings were taken automatically every 50 min for 92 h under conditions of continuous agitation at 37 °C, using four and three separate samples for the acetylated and nonacetylated aS, respectively. The amount of data available so far remained limited by the amounts of available protein and showed some spread in both the lag time and growth rate of the fibrils (Figure S5 of the Supporting Information). However, these data point to the absence of any significant impact of N-terminal acetylation on either the lag phase or the fibril elongation rate. Indeed, under our conditions, both acetylated and nonacetylated aS showed an ~20 h lag phase and an ~22 h fibril elongation phase. The absence of any effect on fibrillation could not be anticipated considering that mutations quite close to the N-terminus of the nonacetylated protein, such as V16P, have been reported to inhibit aS fibrillation substantially. On the other hand, this absence of any effect appears to be consistent with the notion that the structural changes induced by acetylation are very remote from the hydrophobic NAC region of aS (residues 61–95), the main region implicated in aS fibrillation.

### CONCLUDING REMARKS

The vast majority of the very wide array of biophysical measurements reported to date on the structural characterization of aS as well as its interaction with lipids have been taken on bacterially expressed protein that lacks the N-terminal acetylation, a post-translational modification that is present to a very high degree in mammalian cells. An initial report on mammalian-expressed aS suggested the protein exists as an α-helical tetramer, finding subsequently attributed to its N-terminal acetylation but disputed by a number of other research groups.

All of the work described above involving full-length protein was performed on aS that underwent a heating step during purification. We tried to replicate the procedure followed by Trexler and Rhoades but failed to isolate the putative tetrameric form of the protein, possibly because of difficulties in separating it from a host of other proteins. Although the isolated material showed considerable α-helical character as judged by CD, it was heterogeneous in mass and showed precipitation after the sample had been heated to 90 °C, characteristic of irreversible denaturation. To work around these difficulties with aS purification under nonnaturing conditions, we resorted to adding a C-terminal His tag to aS. Using this new construct, we were able to perform gentle purification and obtain highly pure N-terminally acetylated protein (see Materials and Methods for details). The resulting protein showed the same dynamic disorder that was seen for nonacetylated or acetylated monomeric protein, despite the absence of a heat denaturation step and very mild conditions used during purification. When using carefully matched buffer conditions, the only differences between the NMR spectra of the gently purified acetylated His-tagged aS and acetylated WT aS, lacking the His tag but purified using heat denaturation, are localized to the C-terminal residues, presumably reflecting the effect of a weak electrostatic interaction between the mostly positively charged His tag (at pH 6.0) and the mostly negatively charged C-terminal residues (Figure 7). It is conceivable that the C-terminal His tag prevents the formation of the putative tetramer. However, this scenario seems unlikely because in the proposed structure of the tetramer the C-terminal region of aS remains disordered.

In the absence of lipids, the N-terminal acetylation of aS induces a modest ~17% population of α-helical conformation for the first six residues. However, the exchange between α-helix and random coil takes place on a time scale faster than ~1 μs, as judged by the absence of significant line broadening for these N-terminal residues. In the absence of lipids, the chemical shifts of only the very N-terminal residues of aS are visibly impacted by the acetylation and no significant perturbations can be seen in the NMR spectrum beyond residue 12. Considering that the NMR resonances are exquisitely sensitive to the distribution of backbone torsion angles sampled by the molecule, this strongly suggests the absence of any specific interactions of other aS residues with this transient small N-terminal helical segment. At first sight, the very small, ~1.5% increase in the hydrodynamic radius of aS upon acetylation might appear to contradict the absence of long-range interactions. However, it appears that the 20% lower hydrodynamic radius of aS compared to a total random chain of the same length simply reflects a small electrostatic compaction caused by the opposite net charge between the 40 C-terminal residues and the 60 N-terminal residues. This charge difference, and thereby the compaction, is reduced by the removal of one elementary positive charge upon acetylation of the N-terminal amino group.
The distinct increase in lipid affinity seen for N-terminally acetylated aS, which extends well beyond the 20 N-terminal residues as judged by both CD and NMR (Figures 3 and 6), is perhaps surprising. In fact, one might have expected that the removal of the N-terminal positive charge associated with acetylation of the amino group would slightly decrease the affinity of aS for negatively charged lipid surfaces. Instead, we observed that the minor change in the protein’s chemical structure upon N-terminal acetylation, and its associated transient ~17% α-helical character of the first few residues in free solution, strongly impacts the lipid binding properties of the entire protein. This observation points to a special and essential role for the N-terminus of the protein in initiating membrane binding, prior to elongation of the bound section of the protein beyond its dozen immediate N-terminal residues. This “initiation—elongation” model of binding is consistent with the stepwise decrease in NMR resonance attenuation seen in the presence of small quantities of SUVs when moving farther from the N-terminus and explains the strong increase in the level of lipid binding upon acetylation. Indeed, with the helical structure of aS in the lipid-bound state being well established, preformation of helical structure at the N-terminus in aqueous solution will significantly increase the on rate for binding of aS to lipid, while not significantly affecting the off rate. This kinetic change therefore results in an increase in lipid binding affinity. The distinct steps in the attenuation profile as a function of distance from the N-terminus (Figure 6) point to significant kinetic barriers between aS free in solution and the cases where either just the first dozen residues are lipid-bound or the entire N-terminal 100-residue region of the protein interacts with the membrane. A special role in initiating lipid binding has already been ascribed to the N-terminal residues. Our current data for acetylated aS amplify this role considerably and suggest that N-terminal acetylation of the protein is critical for the study of its interaction with lipid vesicles.

**ASSOCIATED CONTENT**

**Overlay of 800 MHz HSQC spectra of nonacetylated and acetylated aS; plots of the differences between chemical shifts measured for nonacetylated aS and random coil values predicted on the basis of the work of Kjaergaard et al. for five nucleic acid polymers, comparison of small expanded regions of the 900 MHz NOESY spectra of the nonacetylated and acetylated 15-residue N-terminal fragment of aS; CD spectra of nonacetylated and acetylated 15-residue peptides; changes in thioflavin T fluorescence as a function of time in the course of fibrillation experiments for nonacetylated and acetylated aS; overlay of 500 MHz HSQC spectra of acetylated wild-type aS and acetylated His-tagged aS, purified under soft conditions; tables of 13Cα, 13Cβ, 15N, and 1H NMR chemical shift assignments measured for acetylated and nonacetylated aS at 15 °C and pH 6; a table of chemical shifts predicted on the basis of the work of Kjaergaard et al. for WT aS at 15 °C and pH 6; and a table of 1H-1H coupling constants for the first 15 residues of acetylated and nonacetylated aS and the average 1H-1H values for the residue types involved. This material is available free of charge via the Internet at http://pubs.acs.org.
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