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This article has been accepted for publication and undergone full peer review but has not been through the copyediting, typesetting, and proofreading process, which may lead to differences between this version and the official version of record.

Please cite this article as: Myagmargerel B, Khulan B, Gantsetseg B, Khongorzul B, and Tuya M. Synthetic fatty acid from crude oil of Tamsagbulag petroleum deposit (Mongolia). Mongolian Journal of Chemistry, 22(48), 2021, xx-xx
doi.org/10.5564/mjc.v22i48.1645
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ABSTRACT
In this work, we studied to obtain synthetic fatty acids raw materials to produce surfactants and various detergents from paraffin. The solid paraffin extracted from the Tamsagbulag petroleum deposit was 46 mass%, distilled 350-450°C, by complexing with urea. Experimental study indicates paraffin’s physicochemical characteristics from this petroleum are more suitable to use in the production of synthetic fatty acids. By solid paraffin oxidation, we obtained 33.01% of the synthetic fatty acids containing monocarboxylic acids.

We suggest obtaining the synthetic fatty acid by oxidation process in normal condition from petroleum paraffin.

Keywords: synthetic fatty acid, solid paraffin, crude oil, oxidation, petroleum
INTRODUCTION

Crude oil, one of the most complex chemical mixtures in the world, serves as a feedstock for the petrochemical industry and as raw material for many chemical products such as fuels, solvents, lubricant oils, plastics, fertilizers, detergents, and others [1, 2].

There are some deposits of crude oil in our country. Out of them, 3 large deposits are being exploited. The Tamsagbulag deposit is located in the Matad area of Dornod province, Mongolia. Other deposits such as Zuunbayan and Tsagan-Els are situated in the Zuunbayan area Dornogovi province, with 332.67 million tons of proven oil reserves and 43.25 million tons of proven reserves [3-5].

Previously, we studied the physicochemical properties of petroleum and the detail of the complete hydrocarbon composition of crude oils from the Tsagaan-Els, Zuunbayan, and Tamsagbulag deposits. Based on the previous report, the crude oil of Mongolia was determined as heavy oil with high viscosity and has a lower yield of light fraction, and it belongs to the high-paraffinic category of crude oil [3, 6, 7]. One of a kind products obtained from the high paraffin crude oil is a synthetic fatty acid. Synthetic fatty acids are critical raw materials for soaps, detergents, fragrances, emulsifiers, organic lubricating oils, and surfactants, which can be obtained by oxidation of paraffin-rich crude oil. During the oxidation process, 70% of the high molecular monocarboxylic acids can be produced by the oxidation of solid paraffin to manufacture aliphatic alcohols and surfactants, and the remaining 30% consume for lubricants (15%) and in other industries [1, 8, 9].

For over some years, various researchers have studied the air oxidation of paraffinic hydrocarbons to produce synthetic fatty acids [1, 8]. In the previous research, based on the high-paraffin crude oil in the field of Tsagaan-Els, solid paraffin was extracted from the oil fraction and further oxidized by molecular oxygen the in the presence of a catalyst to produce synthetic fatty acids [10]. Tsagaan-Els crude oil deposit has a lower reserve than the Tamsagbulag oil deposit. Moreover, it contains low asphaltene, low oil, and high water content than the Tamsagbulag oil deposit [10]. Since the Tamsagbulag and Tsagaan-Els deposit crude oil show different physical and chemical characteristics, it's evident that the technological scheme for obtaining synthetic fatty acid is expected to be different.

At the same time, the development of petrochemical synthesis technology from various petroleum sources in Mongolia is an important study. It has practical significance in creating opportunities for domestic goods production.

In this work, we aimed at the possibility of isolating synthetic fatty acid from solid paraffin from the Tamsagbulag crude oil deposit and studied its properties.
EXPERIMENTAL

Materials: As a research sample of crude oil from the depth of 2300-2600 m in the XXI block of Eastern of Tamsagbulag field Mongolia, which had brought at September 2018 year, was used. The petroleum exploration blocks in Mongolia and sample collected area are shown in Fig. 1.

Experimental procedures: Physical characteristics (density, viscosity, and flash point) and fractional composition were determined according to Petroleum Analysis standard methods [2, 11, 12]. SARA method is used for group composition (oil, resin, and asphaltenes) of crude oil [2, 13, 14]. Experimental to isolate synthetic fatty acids from petroleum were carried out according to the following scheme.

Fig. 1. Petroleum exploration blocks in Mongolia and sample collected area

Fig. 2. Scheme of obtaining fatty acid from Tamsagbulag crude oil
Through forming a complex of urea, solid paraffin was extracted from the oil fraction, which is distilled at 350-450°C, and the structure and properties of obtained solid paraffin have been determined [1, 15, 16].

The oxidation of paraffin was carried out in the liquid phase by molecular oxygen in the presence of catalysts such as permanganates (KMnO₄), temperatures in the range of about 105 to 130°C, and under atmospheric pressure.

The FT-IR spectra of oxide products were obtained on a Nicolet 20-PC FT-IR spectrometer with CsI optics and a DTGS detector. The KBr disc contained a 0.5% finely ground shell sample. All the spectra were measured with a frequency range of 4000 to 400 cm⁻¹ and 32 scans for each sample [17, 18].

The hydrocarbon’s composition of obtained synthetic fatty acids and paraffin of petroleum studied by gas chromatography-mass spectrometer as Thermo Scientific-Trace 1310GC-MS with the detector of TSQ8000-triple Quadrupole MS and column of TR5MS 60x02 mm (Dx0.25 μm) using under 1.5 ml/min speed of helium gas [19-22].

RESULTS AND DISCUSSION

The physical and chemical characteristics, group composition, and content of fraction of the petroleum of Tamsagbulag deposit of Mongolia have been studied. The physical and chemical characteristics of the oil are given in Table 1.

Table 1. The physical and chemical characteristics of Tamsagbulag oil

| Characteristics of crude oil | Tamsagbulag oil |
|-----------------------------|-----------------|
| Appearance                  | thick and black color |
| Specific gravity, 50°C, kg/m³ | 839             |
| Kinematic viscosity, mm²/sec, 50°C | 22.5             |
| Flashpoint, in open crucible, °C | 62              |
| Freezing point, °C          | 18              |
| Group composition, wt %     |                 |
| Resins                      | 10.99           |
| Asphaltenes                 | 0.98            |
| Oil (saturate and aromatics)| 88.03           |
| Solid paraffin in the oil   | 17.8            |
| The yield of fuel fraction, % |                 |
Table 1 shows Tamsagbulag petroleum is a high viscosity and high flash point. The yield of light fractions of petroleum is low, leading to an increase in flash temperature. The high temperature of the flash point of this oil is directly related to the high boiling point. The high paraffin content also leads to high viscosity and freezing points.

From the previous references, high paraffin content crude oils from Karaganda (Kazakhstan), and White tiger (Vietnam) deposits show viscosity at 20°C 13.55, 17.52 mm²/sec, respectively [23, 24]. But, Tamsagbulag crude oil’s viscosity at 50°C was 22.5 mm²/sec. Although Karaganda and White tiger crude oils show near paraffin contents to Tamsagbulag oil, their viscosity varies largely. C_{21}-C_{35} n-alkanes content of the Tamsagbulag oil is considered to be high, while the crude oil from the White tiger contains a low number of C_{21}-C_{35} alkanes and a high amount of light C_{10}-C_{20}, respectively. The high number of C_{20}-C_{34} alkanes of the Tamsagbulag crude oil was caused by its high viscosity [23, 25]. The viscosity, flash temperature, and fuel fraction of the Mongolian other petroleum deposits (Zuunbayan and Tsagaan-Els) were shown the same trend as the Tamsagbulag deposit [3, 6-7, 26]. In other words, Mongolian crude oils are having high density, high flash point, and low fuel fraction than other foreign deposits with the same paraffin content.

The oil (lub) fraction’s content was determined as 30.62% in Tamsagbulag oil, and the oil fraction was viscous and mainly composed of hydrocarbons of paraffin. Therefore, chemical synthetic products can be produced based on the raw materials of the above oil fraction and are economically effective. Urea forms a crystalline complex with alkanes with at least 6 carbon atoms (>C_6) in straight-chain paraffin or more hydrocarbon atoms obtained at room temperature and under normal pressure. Straight-chain alkanes are dominated in the petroleum oil fraction (300-400°C) and have a high ability to form urea complexes under normal conditions [27, 28].

According to our results, iso-octane is more selective in the process of forming a complex with urea in the oil fraction. The best conditions for separating paraffin can be described
as the oil fraction: iso-octane 1:5, oil fraction: urea 1:4, urea: activator (ethanol) 1:4 [27, 28]. In this case, the yield of solid paraffin separated from the oil fraction of the Tamsagbulag deposit was 46.0%. The characteristics of solid paraffin from the oil fraction of the Tamsagbulag deposit have been shown in Table 2.

Table 2. Characteristics of solid paraffin extracted from Tamsagbulag oil

| Characteristics                        | Solid paraffin | Petroleum paraffin C type, GOST 23683 |
|----------------------------------------|----------------|---------------------------------------|
| Appearance                             | White color and hard | White and hard                        |
| Odor                                   | odorless         | odorless                              |
| Yield, %                               | 46.0            | -                                     |
| Specific gravity, g/cm³                 | 0.84            | -                                     |
| Melting point, °C                       | 52              | 45-52                                 |
| Freezing point, °C                      | 44.2°C          | -                                     |
| Temperature, °C                         |                 |                                       |
| 5% distillation temperature, °C        | 355             | >320                                  |
| Distillation up to 400°C, %            | 60              | >60                                   |
| Distillation up to 415°C, %            | 80              | >80                                   |
| 97% distillation temperature, °C       | 445             | >460                                  |
| Number of carbon atoms                  | C₂₀ - C₃₄       | -                                     |
| Content of n-alkanes, %                | 83.12           | -                                     |
| Content of oil, %                      | 2.2             | <2.3                                  |

Table 2 shows that the physicochemical characteristics of paraffin wax obtained from the oil fraction of this oil are classified as C-type paraffin, which can be used to produce synthetic fatty acids [27, 29].

The content and distributions of the n-alkanes composition of paraffin, separated from Tamsagbulag's oil, were determined by gas chromatography and are shown in Figs. 3, 4.
The paraffin obtained from Tamsagbulag oil contains \( n \)-alkanes of 83.12% with carbon atoms \( C_{20}-C_{34} \). If the purity of paraffin is insufficient and the branched alkane content is high, the oxidation process of paraffin to fatty acid is adversely affected. German scientists investigated the solid paraffin of petroleum \( (C_{20}-C_{35}) \) oxidized into fatty acid from lauric acid to stearic acid. Therefore, we used this method which is more suitable to get fatty acid into the general raw material of synthetic soup, washing detergent [1, 31]. Figure 4 specifies that the paraffin obtained from Tamsagbulag deposit oil is a suitable raw material for the production of synthetic fatty acids.

Paraffin molecule oxidation with oxygen and obtaining synthetic fatty acids is a complex reaction controlled by a radical chain reaction mechanism in the liquid phase [1, 31].
Obtaining synthetic fatty acids with the reaction of paraffin oxidation depends directly on
the reaction parameters such as temperature, pressure, catalyst, and oxygen flow. When
the oxidation of paraffin is carried out at temperatures above 120°C, it forms large
quantities of low molecular weight fatty acids -oxyacids, dicarboxylic acids, and other
related products like a mixture various oxygen compounds [32, 33].

Under suitable conditions for synthetic fatty acids, the product yield formed high, and the
by-product is less developed. Therefore, it is necessary to select the appropriate conditions
and carry out the reaction.

In determining the best oxidizing condition of paraffin, the catalyst value was changed
between 0.3-1.5%, oxygen flow 0.5-4.5 l/min, temperature 105-160°C [10, 33].

As the result of oxidation of paraffin of Tamsagbulag’s oil is tested several times in the
temperatures range of about 105 to 130°C with a 0.6 l/min oxygen flow in the oxidizing
reactor, as the following conditions are selected as the best. It is considered that the
amount of catalyst will be 0.3 mas. % [1, 10, 34].

In the early stages of the oxidation of paraffin, we carried out the reaction at relatively high
temperatures (120-130°C) for 2 h to form free radicals and stimulate the faster reaction.

During the process, the catalyst dissolves in paraffin and stimulates an oxidation reaction.

In the second stage of the reaction, the temperature was gradually down to 105-110°C,
and 0.6 l/min flow rate of molecular oxygen was entered, and then an oxidation reaction
takes place. During this process, the paraffin-free radicals are oxidized by oxygen, the by-
product is minor at relatively low temperatures, and the reaction takes place in the direction
of the formation of fatty acids [33, 34].

As a result of the oxidation of n-paraffin, a mixture of acidic and neutral oxygenated
compounds is formed. The amount of these compositions is controlled by the value of acid.

Generally, the value of acid varies from 70 to 76 mg KOH/g when synthetic fatty acid is
oxidized [1, 31].

It shows that the number of oxidizing acids was 74 mg KOH/g when the solid paraffin of
petroleum of Tamsagbulag is oxidized under these conditions for 30 h, and paraffin’s
hydrocarbons are oxidized to form fatty acids.

The dependence between the oxidation period of paraffin and the number of acids is shown
in Fig. 5. Oxidation is high as the number of acids increases, and it is linearly related to the
oxidation period.
Fig. 5. Dependence of value acid and oxidation time of solid paraffin

The structure of the hydrocarbons in the composition of paraffin and oxidized products of Tamsagbulag’s oil was determined by IR spectroscopy and are shown in Fig. 6.

IR analysis for both the paraffin and oxidized paraffin showed characteristic bands at 2850-2916, 1467, and 720 cm\(^{-1}\). The bands are CH-aliphatic, \(-\text{CH}_3\)– methyl, and methylene-(CH\(_2\)). The IR spectrum of the oxidized product shows the intensive absorption of a carboxylic acid group (\(-\text{R-COOH}\)) at 1710 cm\(^{-1}\) [17, 18].

IR spectral analysis confirmed that the paraffin obtained from Tamsagbulag’s oil had been oxidized by the flow of oxygen in the presence of a catalyst to the fatty acid.
The low molecular acids were separated from other mixtures by washing hot distilled water and saponified by Na$_2$CO$_3$ and alkali solution, distilled under 140-260°C/2 mm Hg pressure. Generally, after synthetic fatty acid is formed, the raw material of fatty acids (C$_{10}$–C$_{20}$) is used to manufacture soap [1].

The ester value is between 5-10 mg KOH/g, and the saponification value is 245-260 mg KOH/g, and the acid value is 245-260 mg KOH/g [1, 31]. The parameters of the extracted fatty acids were determined and shown in Table 3.

Table 3. The property of fatty acid obtained from Tamsagbulag oil

| Properties                        | Fatty acid |
|-----------------------------------|------------|
| Yield, %                          | 33.01      |
| Acid value, mg KOH/g              | 254        |
| Saponification value, mg KOH/g    | 260.0      |
| Ester value, mg KOH/g             | 6.2        |

From the Table 4, according to the parameters of synthetic fatty acids obtained by oxidizing petroleum paraffin from the Tamsagbulag deposit, the acids value is 254 mg KOH/g, the esters value is 6.2 mg KOH/g, and the saponification value is 260.0 mg KOH/g. It indicates that fatty acids are formed as a result of the oxidation process.

The chemical composition of the fatty acid obtained by paraffin oxidation was determined by gas chromatography, and the results are shown in Fig. 7 and Table 4.

Fig. 7. GC analysis of synthetic fatty acids of oxidized paraffin of Tamsagbulag oil
Table 4. Chemical composition of fatty acid from paraffin of Tamsagbulag oils

| Fatty acids           | Chemical formula | Content, % | Retention time, min |
|-----------------------|------------------|------------|---------------------|
| Enanthic acid         | C_6H_{13}COOH    | 0.24       | 7.15                |
| Caprylic acid         | C_7H_{15}COOH    | 1.1        | 9.07                |
| Pelargonic acid       | C_8H_{17}COOH    | 2.1        | 10.92               |
| Capric acid           | C_9H_{19}COOH    | 2.74       | 12.68               |
| Undecanoic acid       | C_{10}H_{21}COOH | 3.15       | 14.36               |
| Lauric acid           | C_{11}H_{23}COOH | 3.47       | 15.94               |
| Tridecyl acid         | C_{12}H_{25}COOH | 3.41       | 17.44               |
| Myristic acid         | C_{13}H_{27}COOH | 3.33       | 18.86               |
| Pentadecanoic acid    | C_{14}H_{29}COOH | 3.32       | 20.22               |
| Palmitic acid         | C_{15}H_{31}COOH | 2.97       | 21.51               |
| Margaric acid         | C_{16}H_{33}COOH | 2.54       | 22.75               |
| Stearic acid          | C_{17}H_{35}COOH | 1.9        | 23.93               |
| Nonadecyl acid        | C_{18}H_{37}COOH | 1.36       | 25.06               |
| Arachidic acid        | C_{19}H_{39}COOH | 0.88       | 26.14               |
| Heneicosylic acid     | C_{20}H_{41}COOH | 0.5        | 27.19               |

The total content of fatty acids (C_6–C_{20}) 33.01
Content of non-oxidized paraffin (C_9–C_{23}) 49.45
Other compounds 17.54

As shown in Table 4, the composition of synthetic fatty acids obtained by the oxidation of petroleum paraffin contains synthetic fatty acids with carbon atoms C_7–C_{20} with a variation of content 0.2-3.5%.

In a mixture of synthetic fatty acids extracted from petroleum paraffin at the Tamsagbulag deposit, the monocarboxylic acids are 33.01 and 49.45% of non-oxidized paraffin hydrocarbons, and the content of other oxygen compounds is 17.54%.

As a result of this study, it has been determined that the oxidation of petroleum paraffin at an oxygen flow rate of 0.66 l/min at a variable temperature of 130-105°C, with the presence of 0.3% catalyst (KMnO_4), is the optimal condition of reaction.

Although it has been considered that the paraffin's hydrogen carbons are inactive and require a unique condition for the chemical synthesis, the present research indicated that
the paraffin from the Tamsagbulag oil can be processed to form synthetic fatty acid without using special requirements.

CONCLUSIONS
It has been determined that 46% paraffin can be obtained by forming a complex of urea, solid paraffin extracted from the oil fraction of Tamsagbulag oil distilled at 350-450°C, and it belongs to the technical category C type used in the production of synthetic fatty acids. Our study has indicated that the superior fatty acid, which is a raw material for surfactants, can be obtained by oxidizing petroleum paraffin hydrocarbons with oxygen under normal conditions of the oxygen flow rate of 0.6 l/min at a variable temperature of 130-105°C, with the presence of 0.3% catalyst (KMnO₄). The obtained fatty acid’s total content of fatty acids (C₇-C₂₀) is 33.01%, and the content of non-oxidized paraffin C₇-C₂₀ is 49.45%.

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