Application of Waste Plastic Oil as Alternative Fuel

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Abstract - Pyrolysis of plastic waste had become an important method to produce plastic waste oil as new supply of energy and alternative fuel system. This work was aimed to characterize the plastic waste oil resulted from pyrolysis. The use plastic waste oil was then used for two different applications.

Collected pyrolysis products using different reactor set up were analyzed and tested using GCMS, density, viscosity @ 40 0c, flash point, furan and dioxin and other test done by university of medical science, petroleum laboratories research and studies (PLRS) in Sudan, agricultural research center in Egypt and Egyptian petroleum research institute (EPRI).

Then the final product was used for fueling Tuk Tuk and the performance was found to be comparable to traditional fuel and even better than traditional fuel for TUK TUK as noticed that there a difference of 2 Km when using 600 ml. Adding small amount of domestic oil to PP waste oil and reused again in Tuk Tuk gives excellent result. As increased the distance travelled by one compared to PP oil alone.

All tests results shows good results and confirm that the produced oil from HDPE, LDPE and PP alone is compared to gasoline specifications. Also the produced oil from mixture (HDPE+LDPE+PP) is compared to diesel specifications.

Key Words: Waste Plastic, Conversion, GCMS.

INTRODUCTION:
Waste plastic are one of the most promising resource for fuel because of its high heat of combustion and due to the increasing availability in local communities. Both plastics and petroleum derived fuels are hydrocarbons that contain the element of carbon and hydrogen. The difference between them is that plastic molecules have longer carbon chains than those in LPG, petrol and diesel fuel.

The different waste plastics were thermally cracked at different temperatures and then it was tried to measure the oil produced.

The conversion method of waste plastic into fuel depend on the types of plastic to be targeted and the properties. The type of plastic and their composition will condition the conversion process and will determine the combustion temperature for the conversion, the energy consumption required and the fuel quality out put.

MATERIAL AND METHODS

2.1 Material:
High density polyethylene (HDPE), polypropylene (PP), low density poly ethylene (LDPE), mixed of (500g HDPE + 500 g LDPE) and mixed plastic (250 g HDPE+ 250 g LDPE + 500g PP). all the plastic material were shredding before using.

2.2 Equipment:
Reactor made from iron with different accessories, gas stove, jar, temperature watch and pipe.

2.3 Methods:
Vertical reactor V1:
1. Started the first trial at 11:20 am by adding 1Kg of HDPE until 3:50 pm, the product was approximately 70 ml with plastic remained unconverted shown in fig (3.1).

Figure (3.1): unconverted plastic

2. Putting 850g of HDPE at 11:45 am, after half an hour the bubbles begin at the surface of water then the white gas observed after an hour and 14 minutes. The white gas automatically started and condense to oil in jar of water. After 2 hour the percentage of oil was little due to the degree of hotness and thickness of reactor, increasing the degree of hotness increased the gas subsequently, but the water became too hot so ice was added to help in cooling. An hour later the oil amount increased quickly. The procedure was stopped after 3 hours at 2:45 pm, then separated the oil from water by making a small hole at the bottom of jar allowing the water to pass out and the oil remain in the jar then put it in the glass flask shown in fig (3.2).
3. Repeat the process for anther 1 Kg of HDPE and 1 Kg of PP.
Vertical reactor with one watch V6
1. 1kg of PP run about 1 hour from 11 am to 12 pm, the final product was green oil from 170°c to 380°c.
2. Repeat the process with the same material 1kg of PP and same condition to check the unit doing well. the product was also green oil about 950 ml in same temperature where the percentage of conversation was best one.
3. 1 kg mixed of PP and HDPE equally started at 12 pm to 12:45pm, temperature in range 150°c to 350°c, the final product was almost 950 ml light green oil shown below in fig (3.3).

4. 1 kg of mix LDPE and HDPE equally added to the unit started at 2:45 pm ended at 3:20 pm, the gases started at 150°c to 350°c when the gases complete out the temperature also decrease gradually to know the processed done. where the result is pure and light yellow oil in spite of wax due to LDPE shown in fig (3.4) down. Separate the oil in glass and put in the office.

5. 1 kg mixed of 500 gm of LDPE plus HDPE and 500 gm of PP where added to unit at 11:45 am to 12:30 pm at same temperature, the result was 900 ml green light oil due to the color of PP. We notice in this process increasing of the white gas coming so the losses in condensation bigger than others. Shown in fig (3.5)

- After i finished from every process and separate the oil, put it in the glass take it to the office for keeping until tested. When the oil exposed to cold air from A/C started to turn to wax completely in run (c) , and some blocks and heavy dense oil in run (d) shown in figures below (3:6) and (3:7).
- I decided to return the wax and heavy viscous oil to the unit and running again to see what happen if we had pyrolysis in two stages.
- First we put PP oil from previous processed in unit V6 and run it at 11 am to 11:30 am where the green heavy oil turn to very very light oil and the color was also very light green shown in figure below (3:7). The HDPE wax from previous processed put in the unit and run again for an hour gives light oil. Figure below (3:8).
Testing plastic oil:

The following tests were conducted according to the relevant standards as specified.

Plastic pyrolysis oil (liquid samples)
1) GCMS for PP oil sample done by university of medical science and technology.
2) Density at 15\textdegree c using ASTM D4052 (g/cm\textsuperscript{3}).
3) Kinematic viscosity at 50\textdegree c using ASTM D445-12
4) Flash point using ASTM D93-12 \textdegree c
5) Gross calorific value using ASTM D4868-00 or calculation (Mj/Kg)
6) Viscosity @ 40\textdegree c using ASTM D445 cSt
7) Water by distillation using ASTM D95 (V/V)
8) Micro Carbon residue using ASTM 4530 (Wt%)

All the above tests done by petroleum laboratories, research and studies (PLRS) in Sudan.

Application of the plastic waste oil:

Three trials were made to apply the plastic waste oil as fuel to Tuk Tuk, these were as following:

I. 500 ml of the resulting HDPE oil was used as fuel to replace the traditional fuel of tuk tuk. And the distance travelled by the plastic oil were 13 Km and the domestic fuel were 10 Km. Fortunately, tuk tuk traveled distance 1.3 times the distance travelled by same amount of traditional fuel.

II. Also another 600 ml of PP oil used in tuk tuk travelled times the distance travelled by same amount of traditional fuel. Which was less than the first trial due to the density of it was less than above oil where the burning characteristics was fast.

III. Small amount of domestic oil used to be added to benzene was added to PP oil and reused again in tuk tuk the result being very excellent. Where the distance differ between PP oil plus oil and the PP oil only 1 Km, and the PP oil and traditional fuel for tuk tuk were 2 Km.

IV. 100 ml of mixed plastic oil (HDPE, LDPE, PP) where adding to the cresset and lighting for 9 hours.

RESULTS AND DISCUSSIONS

Table (4.1): of Conversion percentages:

| Material                | Reactor type | Conversion percentage |
|-------------------------|--------------|-----------------------|
| 850 Kg OF HDPE          | Vertical reactor V2 | 695.8/850 *100 =81.7\% |
| 1 Kg of HDPE            | Vertical reactor V2 | 950/1000*100 =95 \% |
| 1 Kg of PP              | V2           | 822.7/1000*100 =82.2\% |
| 1 Kg of PP              | V6           | 950/1000*100 =95\% |
| 1 Kg of mixed (PP and HDPE) | V6        | 950/1000*100 =95\% |
| 1 Kg of mixed (HDPE and LDPE) | V6    | 960/1000*100 =96\% |
| 1 Kg of mixed (HDPE, LDPE and PP) | V6    | 900/1000*100 =90\% |

The result of GCMS for PP oil sample done by university of medical science and technology was C7 to C13, which in the range of benzene.
Table (4.2): Test of sulfur, lead content, dioxin and furan content:

| Sample ID | Sulfur content mg/l | Lead content mg/l | Dioxin and furan (PCDD/PCDF) |
|-----------|---------------------|-------------------|-------------------------------|
| HDPE oil  | Nil                 | > 0.375           | 0.18                          |
| PP oil    | Nil                 | > 0.365           | -                             |

- This result done in Egypt which was being very difficult to do again for other running experiment. For PCDD/PCDF the measurement uncertainty expressed as expanded uncertainty (at 95% confidence level) is within the range +50%.

Table (4.3): The tests result was for PP oil and HDPE oil run from V1:

| Sample ID | Density @ 15°C g/cm³ | FT-IR          |
|-----------|-----------------------|----------------|
| HDPE oil  | 0.7884                | According to the scan, saturated (-C-H) stretching peaks were found, (C=C) were found, methyl group (-CH3) was found and methylene group (-CH2-) also was noticed. |
| PP oil    | 0.7781                | According to the scan, saturated (-C-H) stretching peaks were found, (C=C) were found, methyl group (-CH3) was found and methylene group (-CH2-) also was noticed. |

Table (4.4): test

| Test name  | Unit                    | HDPE oil pyrolysis twice | PP oil pyrolysis twice | HDPE + LDPE oil pyrolysis twice | HDPE+LDPE+PP oil pyrolysis twice |
|------------|-------------------------|--------------------------|------------------------|----------------------------------|----------------------------------|
| Flash point | °C                      | < 40                     | < 40                   | 60                               | < 40                             |
| Density    | g/cm³                   | 0.7968                   | 0.7739                 | 0.7966                           | 0.7883                           |
| Calorific value gross Mj/Kg | 44.838 | 45.039 | 44.847 | 44.912 |
| Viscosity  | cST                     | 3.173                    | 1.508                  | 2.826                            | 2.020                            |
| Water content | V/V   | 0.00                     | 0.00                   | 0.00                             | 0.00                             |
| Carbon residue | Wt %  | 0.4                      | 0.1                    | *                                | *                                |
| Sulfur content | %m   | *                        | *                      | 0.122                            | 0.071                            |

- Sulfur content test for HDPE oil and PP did not done due to technical reasons at that time. Also MCR test for HDPE+LDPE oil and mixed oil did not done.

CONCLUSION

- From the experimental studies it is clear that the waste plastic oil is a good and also can be used.
- The product can used as fuel for TUK TUK which was better from traditional benzene and more efficiency.
- Also can used as a source of lighting for traditional cresset for rural areaes.

RECOMMENDATION

- More tested like CO, CO2 and NOx concentrations in the plastic oil.
- Poly ethylene terephthalate (PET) and tires should be used in future work studies.

REFERENCE

[1] United Nations Environmental Programme, (2009) “converting waste plastic into a resource”, compendium of technology, Japan.
[2] http://www.astm.org/Standards/D4052
[3] http://www.astm.org/Standards/D445-12
[4] http://www.astm.org/Standards/D93-12
[5] http://www.astm.org/Standards/D4868-00
[6] http://www.astm.org/Standards/D445
[7] http://www.astm.org/Standards/D95
[8] http://www.astm.org/Standards/D4530
[9] http://www.astm.org/Standards/D4691
[10] Conversion of waste plastics into alternative fuel, "Rao*, 4.(8): August, 2015", “Dr.L.Nageswara Rao*, J L Jayanthi , D.Kamalakar”
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