Phytochemical screening and in vitro evaluation of antioxidant and antibacterial activities of *Teucrium trifidum* crude extracts

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ABSTRACT

The objective of the investigation was to identify biologically active polyphenols and to determine the antioxidant and antimicrobial capacity of *Teucrium trifidum* extracted with different organic solvents (acetone, ethanol and methanol) and distilled water. The results of the study revealed varying levels of polyphenols in the different solvent extracts. Condensed tannin, flavonoid and total phenolic content ranged from (77.339 ± 1.068) to (99.395 ± 1.490) mg CE/g; (3.398 ± 0.2416) to (53.253 ± 0.638) mg QE/g; (14.1087 ± 0.0915) to (21.7977 ± 0.0279) mg GAE/g, respectively. The extracts demonstrated high antioxidant activity in 2,2-azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS), 2,2-diphenyl-1-picrylhydrazyl (DPPH), nitric oxide (NO) and total antioxidant capacity (TAC) assays which were comparable to rutin and butylated hydroxytoluene (BHT) and increased with increasing concentrations of polyphenols extract (P < 0.05). The agar dilution assay of acetone, ethanol and methanol extracts revealed an appreciable broad-spectrum activity against tested pathogenic bacteria. The findings of this study provide evidence that *T. trifidum* can be used as a natural source of antioxidant and antimicrobial components.

1. Introduction

One of the main actions taken by the meat industry to extend the shelf-life and quality of meat and meat products is through the use of meat preservatives (Lorenzo et al., 2018). These meat preservatives prevent oxidation, enzyme reactions and microbial spoilage, which impair the nutritional value and quality of meat and meat products (Falowo et al., 2014). Synthetic antioxidants and antimicrobials have been and continue to be used in the meat industry as preservative agents to reduce economic losses and protect consumer health. However, there has been a serious concern about the use of synthetic preservatives in the food production chain due to the adverse effects on consumers (Lobo et al., 2010). Importantly, the use of such synthetic antimicrobials in the meat production chain has resulted in the emergence of bacterial-resistant strains that render the management and treatment of livestock, poultry and human diseases difficult (Chellat et al., 2016), thus creating a serious need for novel antibacterial agents.

As a result, the ability of synthetic preservative agents to cause secondary undesirable effects has encouraged consumers and the meat industry to call for natural preservatives (Karre et al., 2013). In addition, consumers are now insisting on high-quality meat and meat products with a long shelf life and no synthetic preservatives (Kumar et al., 2015). The meat industry is therefore currently responding to this call for natural preservatives to meet consumer demands (Karre et al., 2013; Kumar et al., 2015). In recent years, due to their ability as non-toxic preservatives in meat and meat products, considerable attention has been paid to natural preservatives, in particular, antioxidants and antimicrobials of plant origin (Kumar et al., 2015). Plants are an invaluable source of phytochemicals, among them alkaloids, phenolic acids, terpenes, terpenoids, flavonoids, tannins and essential oils (Aziz and Karboune, 2016).
that have bioactive effects, including antioxidant and antimicrobial (Falowo et al., 2014; Şahin et al., 2017).

Several researchers have demonstrated the efficacy of various natural antioxidants and antimicrobial agents with the potential to reduce lipid oxidation, spoilage and microbial load in meat and meat products (Falowo et al., 2017; Ergezer et al., 2018). In addition, plant extracts have demonstrated activity against food-borne and pathogenic bacteria, including Bacillus cereus, Escherichia coli, Klebsiella pneumoniae, Pseudomonas aeruginosa, Salmonella Typhimurium and Staphylococcus aureus (Mostafa et al., 2018). For that reason, the inclusion of natural antioxidants and antimicrobial agents in meat and meat products may increase the shelf-life and enhance the nutritional quality of meat and meat products.

*T. trifidum* Retz. is one of the many plants with phytochemicals exhibiting these beneficial biological activities. The shrub is widely distributed in South Africa. It belongs to the Lamiaceae family and is used as an ethnomedicine in the management of bacterial-induced diseases as well as an ethnoveterinary medicine in gastrointestinal parasites (Maphosa and Masika, 2010); gall sickness (Masika and Afolayan, 2003); digestive and respiratory ailments (Ruiters et al., 2016). Research has also shown that *T. trifidum* essential oil has antibacterial activity against *Staphylococcus aureus* (Ruiters et al., 2016). Despite the well documented antioxidant and antibacterial activity of *T. trifidum*, its polyphenol content, and its potential *in vitro* antioxidant and antibacterial activity have not been determined. This study therefore sought to determine the polyphenol content, antioxidant and antimicrobial activity of *T. trifidum* extracts.

2. Materials and methodology

2.1. Plant collection

The *T. trifidum* shrubs were collected from the natural veld in the village of Mbizana, Eastern Cape, South Africa. The plant was authenticated at the Albany Museum Herbarium in Grahamstown, South Africa (voucher specimen number IMAZ08/2018). The collected shrubs were rinsed with distilled water and air-dried for 7 days. Dry shrubs were ground into powder with a blender (Waring Products Division, Torrington, USA).

2.2. Plant extracts preparation

Four samples of 100 g of *T. trifidum* powder were soaked in 1 L of acetone, distilled water, ethanol and methanol. The resulting mixtures were shaken on an orbital shaker (Stuart Scientific Orbital shaker, UK) for 48 h at room temperature. The crude extracts were then filtered through a pressure filter using the Buchner funnel and Whatman No. 1 filter paper. The acetone, ethanol and methanol extracts were then condensed in a rotary evaporator (Laborator 4000-efficient, Heidolph, Germany), while the aqueous extract was freeze-dried (Vir Tis benchtop K, Vir Tis Co, Gardiner, NY). The extracts were stored at 4 °C in airtight glass bottles.

2.3. Determination of polyphenols

2.3.1. Condensed tannin content

The condensed tannin content was determined following the procedure by Han et al. (2009). Briefly, the reaction mixture contained 0.5 mL (1 mg/mL) of the extract plus 3 mL of vanillin-methanol (4 % w/v) and 1.5 mL of hydrochloric acid. The standard catechin (0.02–1 mg/mL) reaction mixture was also prepared in the same way. The reaction mixtures were then vortexed and left to stand at room temperature for 15 min. The absorbance of the reaction mixtures was then measured at 500 nm using a UV-3000 PC spectrophotometer (J.P. Selecta, Spain). Condensed tannin content was determined by the calibration curve equation: \( y = 0.8462x + 0.244, R^2 = 0.998 \) and expressed as milligram of catechin equivalent (mg CE/g) using the equation:

\[
C = c \times V/m
\]

Where \( C \) = total condensed tannin constituent in mg as CE/g extract, \( c \) = catechin concentration in mg/mL derived from the calibration curve, \( V \) = volume of the extract in the reaction solution in mL, and \( m \) = weight of the extract used in the assay in g.

2.3.2. Flavonoid content

The determination of the flavonoid content was done using the aluminium chloride colorimetric method of Abifaririn et al. (2019) with some modifications. Briefly, 0.5 mL (1 mg/mL) of the different solvent extracts at varying concentrations (0.2–1 mg/mL) were added to the respective test tubes. Subsequently, 2 mL of distilled water and 0.15 mL of 5 % sodium nitrite were added to each of the test tubes and the mixture was left standing for 6 min. The quercetin standard reaction mixtures were similarly prepared. Then 0.15 mL of 10 % AlCl3 was pipetted into the reaction mixture which was then left standing for another 5 min. Then 1 mL of 1 M sodium hydroxide was added, followed by the addition of distilled water to make up a 5 mL solution. Absorbance was read at a wavelength of 420 nm. Flavonoid content was determined from the quercetin calibration curve equation: \( y = 1.968x + 0.1572, R^2 = 0.9856 \) and presented as milligram of quercetin equivalent (mg QE/g) using the equation:

\[
C = c \times V/m.
\]

2.3.3. Total phenolic content

The total phenolic content of acetone, water, ethanol and methanol extracts of *T. trifidum* were determined using the Folin–Ciocalteu reagent (Unuofin et al., 2018). Briefly, 0.5 mL of the plant extracts (1 mg/mL) and standard gallic acid (0.02–0.1 mg/mL) were pipetted into test tubes. Subsequently, 2.5 mL of 10 % (v/v) Folin–Ciocalteu reagent was added and the mixture was vortexed. The vortexed solution was left standing at room temperature for 5 min, followed by an addition of 2 mL of 7.5 % (w/v) anhydrous sodium carbonate. The mixture was then incubated for 30 min in a water bath at 40 °C for colour development. The absorbance was measured at 765 nm. The phenolic content was then calculated using the gallic acid calibration curve (\( y = 27.398x + 0.1099, R^2 = 0.984 \)) and presented as milligram of gallic acid equivalent (mg GAE/g) using the equation:

\[
C = c \times V/m.
\]

2.4. Antioxidant assay

2.4.1. ABTS radical scavenging activity

The ABTS scavenging activity of the plant extracts was determined as described by Unuofin et al. (2018). The reaction mixture was made by adding equal volumes of 7 mM ABTS and 2.45 mM potassium persulfate. The mixture was left standing at room temperature in the dark for 12 h to release ABTS radicals. The resulting green-coloured solution was then diluted by adding 3 mL of the ABTS solution to 150 mL of methanol to get an absorbance of 0.700 ± 0.005 at 734 nm. Once the required absorbance was obtained, 1 mL of the resulting solution was combined with 1 mL of the plant extracts/or standards at varying concentrations (0.005–0.08 mg/mL). Approximately 7 min later, the decrease in absorbance was measured at 734 nm. The percentage inhibition of ABTS radicals by the extracts or standards was determined using the following equation: ABTS Scavenging activity (%) = [(Absorbance of control – Absorbance of the sample)/(Absorbance of control)] × 100.

2.4.2. DPPH radical scavenging assay

The procedure described by Abifaririn et al. (2019) was followed to determine the free radical scavenging activity of extracts on DPPH radical. A mixture of 0.135 mM DPPH radical in methanol was made and 1 mL of this mixture was combined with 1 mL (0.005–0.08 mg/mL) of each extract. The reaction mixture was allowed to stand The procedure
described by Abifarín et al. (2019) was followed to determine the free radical scavenging activity of extracts on DPPH radical. A mixture of 0.135 mM DPPH radical in methanol was made and 1 mL of this mixture was combined with 1 mL (0.005–0.08 mg/mL) of each extract. The reaction mixture was allowed to stand at room temperature in the dark for 30 min. Absorbance was then measured at 517 nm. The reduction of the absorbance was recorded against the control. Rutin and BHT were used as standards (0.005–0.08 mg/mL). in the dark for 30 min. Thereafter, absorbance was measured at 517 nm. The reduction in absorbance was recorded against the control. Rutin and BHT were used as standards (0.005–0.08 mg/mL). The scavenging ability of the extracts was determined using the equation:

\[
\text{DPPH scavenging activity (\%)} = \left(\frac{\text{Absorbance of control} - \text{Absorbance of the sample}}{\text{Absorbance of control}}\right) \times 100
\]

### 2.4.3. Nitric oxide scavenging activity

The nitric oxide assay was done as described by Ohikhena et al. (2018). Briefly, 0.5 mL of each extract and standard at varying concentrations (0.025–0.400 mg/mL) was combined with 2 mL of 10 mM sodium nitroprusside made in 0.5 mM phosphate buffer saline (pH 7.4). The resulting solution was then left standing in a water bath at 25 °C for 2.5 h. Thereafter, 0.5 mL of Griess mixture and 1 mL of naphthalene diimide dichloride (0.1% w/v) were added. The Griess mixture consisted of 1 mL sulfanilic acid reagent that is 0.33 % made in 20 % glacial acetic acid. The mixture was left standing in a water bath at room temperature for another 30 min. The absorbance was then read at 540 nm. A solution containing water instead of the extract/standard was used as a control. The number of nitric oxide radicals inhibited by the extracts was extrapolated from the following equation:

\[
\text{(% inhibition of NO)} = \left(\frac{\text{Absorbance of control} - \text{Absorbance of the sample}}{\text{Absorbance of control}}\right) \times 100
\]

### 2.4.4. Total antioxidant capacity (phosphomolybdenum assay)

The TAC of the plant extracts was determined using the phosphomolybdenum assay as described by Ohikhena et al. (2018). Briefly, 0.3 mL of the extracts and standards (0.025–0.4 mg/mL) were pipetted into test tubes and combined with 3 mL of the reaction mixture (0.6 M sulfuric acid, 4 mM ammonium molybate and 28 mM sodium phosphate). The test tubes were then capped and incubated at 95 °C in a water bath for 95 min. Then the solution mixture was left to cool and the absorbance was read at 695 nm. As a control, a mixture consisting of distilled water was used instead of the extract/standard. The percentage of inhibition was determined using the following equation:

\[
\text{% Inhibition} = \left(\frac{\text{Absorbance of sample} - \text{Absorbance of control}}{\text{Absorbance of the sample}}\right) \times 100
\]

### 2.5. Antibacterial activity of the plant extracts

#### 2.5.1. Bacterial strains

The rationale for the selection of microorganisms was the use of bacteria responsible for food poisoning and pathological effects. The bacteria were acquired from the Microbiology Unit of the Botany Department of the University of Fort Hare, South Africa. Strains from the American Type Culture Collection (ATCC) were used. Four Gram-positive bacteria: *Staphylococcus aureus* (ATCC 18824), *Streptococcus pyogenes* (ATCC 19615), *Bacillus subtilis* (ATCC 6633), *Bacillus cereus* (ATCC 10702) and four Gram-negative bacteria: *Pseudomonas aeruginosa* (ATCC 19582), *Klebsiella pneumonia* (ATCC 4352), *Vibrio cholera* (ATCC 14033), *Salmonella Typhimurium* (ATCC 13311) were used to determine the antibacterial activity of *T. trifidum* extracts.

#### 2.5.2. Bacterial inoculum preparation

The procedure of suspension of the colony by the European Committee for Antimicrobial Susceptibility Testing (EUCAST, 2003) was adopted in the preparation of the inoculum. All the test bacteria were cultured on nutrient agar slants, recovered in sterile Muller-Hinton broth (MHB) and incubated at 37 °C for 24 h. The previous cultures were mixed 1:100 v/v with fresh sterile MHB and grown on Mueller-Hinton agar (MHA) overnight at 37 °C. Similar colonies from the culture were dispersed in 0.85 % sterile saline, diluted with saline, and balanced with 0.5 McFarland standards to obtain a suspension density equivalent to 1 x 10^8 CFU/mL. The suspensions were then confirmed by spectrophotometric measurement at a wavelength of 600 nm. Finally, the cell suspensions were adjusted 1:100 by taking 0.1 mL of the bacterial suspension to 9.9 mL of sterile broth to give an estimated inoculum of 1 x 10^6 CFU/mL. Inocula suspensions were used for inoculation in less than 15 min.

#### 2.5.3. Agar dilution assay

The methods described by Wiegand et al. (2008) and the EUCAST (2003) were used in the antibacterial assay. The MHA was heated in an autoclave at 121 °C for 15 min, then left to cool to 50 °C in a water bath. One millilitre of the 2-fold serial dilutions was transferred to the liquefied agar (19 mL) and carefully whirled before pouring into sterile petri dishes and then left to harden. A 10 µL was taken from the bacterial inocula and then dotted on separate hardened agar to obtain the desired absolute inoculum of 1 x 10^4 CFU/mL. The concentrations of extracts used in the antibacterial assays ranged from 0.3125 mg/mL to 5 mg/mL. The extracts were compared against the antibacterial standard, erythromycin (0.001–0.016 mg/mL). The petri dishes with bacteria were incubated at 37 °C for 24 h. The minimum inhibitory concentrations (MICs) of the extracts were considered to be the lowest concentration with visible growth on the agar plates.

#### 2.5.4. Preparation of the extracts

100 mg/mL stock mixture was made in a small quantity of dimethyl sulfoxide (DMSO) and topped up with MHB. A 2-fold serial dilution of the extracts (50, 25, 12.5, 6.25, 3.125, 1.5625 mg/mL) was prepared. Similarly, the standard antibiotic (erythromycin) was prepared.

#### 2.6. Statistical analyses

The data were presented as mean ± standard deviation. MINITAB 17 statistical packages were used to analyze the data. One-way analysis of variance (ANOVA) was done to determine the variations in the antioxidant and antibacterial activity of the different *T. trifidum* extracts. The means were separated by the Fisher’s Least Significant Difference (LSD). Significance was set at \( P < 0.05 \).

### 3. Results

#### 3.1. Phytochemical content

Condensed tannin, flavonoid and total phenolic content in *T. trifidum* extracts are shown in Table 1. The polyphenol content of acetone, aqueous, ethanol and methanol extracts of *T. trifidum* were significantly different \( ( P < 0.05 ) \). The aqueous extract had the most condensed tannin content \( (99.395 \pm 1.490 \text{ mg CE/g}) \) while the acetone extract had the lowest \( (77.239 \pm 1.068 \text{ mg CE/g}) \). The acetone extract had the highest flavonoid content \( (53.253 \pm 0.638 \text{ mg QE/g}) \), while the aqueous extract had the lowest content \( (3.398 \pm 0.241 \text{ mg QE/g}) \). Total phenolic content \( (21.798 \pm 0.028 \text{ mg GAE/g}) \) was highest in the acetone extract, while the aqueous extract had the lowest total phenolic content \( (14.109 \pm 0.092 \text{ mg GAE/g}) \).
3.2. ABTS radical scavenging activity

Figure 1 shows that the ABTS scavenging activity of the extracts and standards had an increasing trend as concentrations increased (P < 0.05). At the maximum concentration (0.08 mg/mL), the highest % mean inhibition was produced by rutin, BHT and ethanol extract (100 %) followed by methanol (97.52 %), acetone (96.71 %) and aqueous extract (69.36 %). The concentration required to scavenge 50 % of the radicals (IC50) was in the following order: rutin > BHT > ethanol extract > methanol extract > acetone extract > aqueous extract (Table 2). The DPPH radical scavenging activities of the extracts and standards increased as concentrations increased (P < 0.05). At the maximum concentration (0.40 mg/mL), the percentage of nitric oxide inhibition was produced by rutin, BHT and ethanol extract (100 %) followed by methanol (97.52 %), acetone (96.71 %) and aqueous extract (39.64 %) followed in decreasing order (P < 0.05). The concentrations (mg/mL) required to scavenge 50 % of the radicals (IC50) was in the following order: rutin > BHT > ethanol extract > methanol extract > acetone extract > aqueous extract (Table 2).

3.3. DPPH radical scavenging assay

The DPPH radical scavenging activities of the extracts and standards are shown in Figure 2. At the highest concentration (0.08 mg/mL), rutin demonstrated the highest DPPH scavenging activity (93.28 %), while acetone extract (92.67 %), ethanol extract (92.15 %), methanol extract (91.65 %), BHT (90.92 %) and the aqueous extract (39.64 %) followed in decreasing order (P < 0.05). The concentrations (mg/mL) required to scavenge 50 % of the radicals (IC50) followed this trend: rutin > BHT > ethanol extract > methanol extract > acetone extract > aqueous extract. The least IC50 of the ethanol extract was indicative of its strong DPPH radical scavenging activity (Table 2).

3.4. Nitric oxide scavenging activity

The percentage nitric oxide scavenging activity of the extracts and standards increased as concentrations increased (P < 0.05). At the highest concentration (0.40 mg/mL), the percentage of nitric oxide inhibitory activity followed this sequence: rutin > BHT > acetone extract > methanol extract > ethanol extract > aqueous extract (Figure 3). The IC50 values of the extracts/standards had an increasing trend as concentrations increased (P < 0.05). At the maximum concentration (0.08 mg/mL), the percentage nitric oxide scavenging activity followed this sequence: rutin > acetone extract > ethanol extract > methanol extract > aqueous extract (Table 2).

3.5. Total antioxidant capacity

As shown in Figure 4, the extracts displayed a concentration-dependent total antioxidant capacity (P < 0.05). At the maximum concentration, the plant extracts showed high antioxidant activity (74.11–91.01 %) which was comparable to the standards (P < 0.05). On the basis of the IC50, rutin showed higher antioxidant activity compared to the rest. Antioxidant activity followed in this order: rutin > methanol extract > ethanol extract > acetone extract > aqueous extract > BHT.

3.6. Antibacterial activity

Both Gram-positive and Gram-negative bacteria were susceptible to the T. trifidum extracts (Table 3). However, the aqueous extract exhibited the lowest antibacterial activity on both Gram-positive and Gram-negative bacteria with MIC values greater than 5 mg/mL. The standard erythromycin exhibited the highest antibacterial activity with MIC values within the range of 0.001–0.008 mg/mL. The acetone extract was active against all the bacterial isolates except B. cereus and S. aureus. All bacteria types were affected by methanol extracts except S. aureus.

4. Discussion

Phytochemicals present in plants exhibit health beneficial bio-activities including antioxidant (Lorenzo et al., 2018) and antimicrobial activities (Mahmud and Khan, 2018). Due to their antioxidant and antimicrobial function, plants can potentially be used as natural preservatives for meat and meat products. Phenolic compounds such as tannins, flavonoids and phenolic acids are the major antioxidant

### Table 1. Polyphenol content of various solvent extracts of T. trifidum.

| Solvent   | Condensed tannin (mg CE/g) | Flavonoid (mg QE/g) | Total phenolic (mg GAE/g) |
|-----------|---------------------------|---------------------|---------------------------|
| Acetone   | 77.339 ± 1.068            | 53.253 ± 0.638      | 21.798 ± 0.028            |
| Aqueous   | 99.395 ± 1.490            | 3.396 ± 0.241       | 14.109 ± 0.092            |
| Ethanol   | 81.156 ± 0.648            | 45.733 ± 0.870      | 20.356 ± 0.046            |
| Methanol  | 83.701 ± 0.245            | 21.084 ± 0.638      | 14.334 ± 0.087            |

Abbreviations: mg CE/g - milligram catechin equivalent per gram of extract; mg QE/g - milligram quercetin equivalent per gram of extract; mg GAE/g - milligram gallic acid equivalent per gram of extract.

Mean with *-d superscripts within a column indicates significant differences (P < 0.05).
The redox properties of these phytochemicals are important for the absorption and neutralization of free radicals, degradation of peroxides, and also for the quenching of reactive oxygen species (Hossain et al., 2011).

The acetone extract of *T. trifidum* had the highest amounts of total phenolic content and flavonoids while the aqueous extract had the highest amount of condensed tannin content. Such results reveal similar patterns in phenolic and flavonoid content of *Vernonia mespilifolia* acetone and aqueous extracts, respectively (Unuofin et al., 2018). Polarity and solvent influence the amount of phenolic substances derived from plants. The variations in the extraction ability of solvents may

| Extracts/Standard | ABTS | DPPH | Nitric Oxide | TAC |
|-------------------|------|------|--------------|-----|
|                   | IC50 | R^2  | IC50 | R^2 | IC50 | R^2 | IC50 | R^2 |
| Aqueous           | 0.297| 0.999| 0.095| 0.966| 0.292| 0.976| 0.121| 0.915|
| Acetone           | 0.086| 0.942| 0.018| 0.951| 0.290| 0.988| 0.044| 0.977|
| Ethanol           | 0.067| 0.942| 0.012| 0.917| 0.150| 0.995| 0.035| 0.836|
| Methanol          | 0.182| 0.944| 0.017| 0.950| 0.304| 0.993| 0.025| 0.968|
| BHT               | 0.019| 0.765| 0.005| 0.903| 1.914| 0.958| 0.222| 0.960|
| Rutin             | 0    | 0.500| 0    | 0.675| 0.100| 0.875| 0.020| 0.948|

**Abbreviations:** IC50 – the concentration (mg/mL) required to scavenge/inhibit 50% of the radical; R^2 – coefficient of determination.

Values obtained from regression curve with 95% confidence level.

![DPPH radical scavenging activity of T. trifidum extracts. A set of bars with different superscripts are significantly different (P < 0.05).](image1)

![Nitric oxide radical scavenging activity of T. trifidum extracts. A set of bars with different superscripts are significantly different (P < 0.05).](image2)

phytochemicals (Shahidi and Ambigaipalan, 2015). The redox properties of these phytochemicals are important for the absorption and neutralization of free radicals, degradation of peroxides, and also for the quenching of reactive oxygen species (Hossain et al., 2011).

The acetone extract of *T. trifidum* had the highest amounts of total phenolic content and flavonoids while the aqueous extract had the highest amount of condensed tannin content. Such results reveal similar patterns in phenolic and flavonoid content of *Vernonia mespilifolia* acetone and aqueous extracts, respectively (Unuofin et al., 2018). Polarity and solvent influence the amount of phenolic substances derived from plants. The variations in the extraction ability of solvents may
The study showed that synthetic antioxidants (BHT and rutin) had the greatest antioxidant effect on almost all the assays. The various extracts displayed significant free radical scavenging activity therefore *T. trifidum* can potentially be used as a natural source of antioxidants. As a result, the significant antioxidant activity demonstrated by *T. trifidum* extracts could be pointed to the presence of these phytochemicals in the extracts. For example, the ethanol extract had the highest ABTS scavenging potential compared to the other extracts. This may be due to the high phenolic content of the extract, which could have contributed to the electron transfer/hydrogen donation (Unuo et al., 2018). The acetone extract showed superior scavenging ability for DPPH and nitric oxide radicals, while the methanol extract had higher total antioxidant capacity. Flavonoids and phenols are known to have antiradical and antioxidant activity (Heim et al., 2002). As a result, the acetone extract could have scavenged best due to its high phenolic content. In addition, the results indicate that *T. trifidum* extracts can potentially be used as natural sources of antioxidants due to their high free radical scavenging capabilities.

Owing to the use of conventional (synthetic) pharmacological agents such as antibiotics and growth promoters of livestock and poultry, there has been an increase in the problem of antibiotic resistance in consumers as well as livestock and poultry (Marshall and Levy, 2011; Ralte et al., 2019). The possible future threat of this antibiotic resistance has led to research into the use of natural alternatives that are considered to be safe. The antimicrobial capacity of plants is assumed to be the function of phytochemical compounds present in them. Which include alkaloids, tannins, saponins, flavonoids and phenolic compounds (Wintola and Afolayan, 2015). Although mechanisms for antimicrobial activity are not well understood, it is argued that these phytochemicals induce antimicrobial activity by specific mechanisms.

One of the postulated mechanisms is that the phytochemicals interact with and disrupt the phospholipid cell membranes, resulting in increased cell permeability, leading to the depletion of cell components (Omojate et al., 2014). Phenolic compounds inhibit hydrolytic enzymes thus disrupting both the energy production and the production of structural components (Pandey & Kumar, 2013). In bacterial cells, phytochemicals cause cell content coagulation and also inactivate deoxyribonucleic acid (DNA) which is manifested as growth retardation (Omojate et al., 2014).

The results of this study have shown that both gram-positive and gram-negative bacteria are susceptible to *T. trifidum* extracts. The observed antimicrobial effects of *T. trifidum* extracts are ascribed to the phytochemical-induced antibacterial activity as the extracts contained bioactive compounds known to have antibacterial activity. The presence of several phytochemicals with antibacterial activity in the extracts may have contributed to the synergistic destruction of the bacteria (Farasat et al., 2015).
et al., 2014). Although the organic solvent extracts showed efficacy, the aqueous extract was less effective mainly due to poor extraction of the key phytochemicals with antibacterial activity. Biswas, Rogers, McLaughlin, Daniels, and Yadav (2013) reported that gram-negative bacteria had resistance to the antimicrobial activity of plant-derived extracts. Nonetheless, in the current research, the T. trifidum extracts showed activity against gram-negative and gram-positive bacteria. A related study, but using extracts from Bidens pilosa and Moringa oleifera (Falowo et al., 2016), also showed potency against gram-negative bacteria. It is noteworthy that like the T. trifidum extracts, the B. pilosa and M. oleifera extracts, also contained phytochemicals with antibacterial activity.

5. Conclusion

The aqueous, acetone, ethanol and methanol T. trifidum extracts contained condensed tannins, flavonoids and phenolic compounds which showed significant free radical mapping capabilities. In addition, the extracts demonstrated antibacterial activity against gram-negative and gram-positive bacteria. The T. trifidum extracts can potentially be used as a natural source of antioxidant with antimicrobial components and can therefore be used as preservatives.

Declarations

Author contribution statement

Irene R. MAZHANGARA, Emrobowanans M. IDAMOKORO: Conceived and designed the experiments; Performed the experiments; Analyzed and interpreted the data; Wrote the paper.

Eliton CHIVANDI: Conceived and designed the experiments; Wrote the paper.

Anthony J. AFLOLAYAN: Conceived and designed the experiments; Contributed reagents, materials, analysis tools or data; Wrote the paper.

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Competing interest statement

The authors declare no conflict of interest.

Additional information

No additional information is available for this paper.

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