Phenolic compounds in water, suspended particulate matter and sediment from Weihe River in Northwest China
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ABSTRACT

The distribution and ecological risks of 11 phenolic compounds were studied in Weihe River, Northwest China. The concentrations of phenolic compounds were determined by ultra-high performance liquid chromatography (UPLC). The total concentration of 11 phenolic compounds ($\sum$PC$_{11}$) ranged from 0.06 to 14.12 $\mu$g/L with an average of 5.22 $\mu$g/L in water, from 0.92 to 34,885 $\mu$g/g with an average of 4,446 $\mu$g/g in suspended particulate matter (SPM), and from 3.54 to 34.09 $\mu$g/g with an average of 11.09 $\mu$g/g in sediment. For individual phenolic compound, the mean concentration of pentachlorophenol was the highest in water (2.65 $\mu$g/L) and in SPM (3,865 $\mu$g/g), while in sediment the mean concentration of 2,4,6-trichlorophenol was the highest (3.05 $\mu$g/g). The total concentration of 5 chlorophenols ($\sum$CP$_5$) was significantly higher than that of 6 non-chlorophenols ($\sum$NCP$_6$) in all three studied compartments. The phenolic compounds in Weihe River were at moderate levels in water and at high levels in sediment. The ecological risk assessment results indicated that phenolic compounds exhibited a high ecological risk in Weihe River water. In most sites, the distribution coefficient (Kd) (SPM) was much higher than Kd (sediment), which probably suggested fresh phenolic compounds input in Weihe River.

Key words | ecological risk, phenolic compounds, sediment, SPM, water, Weihe River

HIGHLIGHTS

- This is the first systematic study of phenolic compounds in rivers in northwest China.
- The results of ecological risk assessment indicated that phenolic compounds exhibited a high ecological risk in Weihe River water.
- The total concentration of 5 chlorophenols ($\sum$CP$_5$) was significantly higher than that of 6 non-chlorophenols ($\sum$NCP$_6$) both in water and sediment.

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GRAPHICAL ABSTRACT

INTRODUCTION

Phenolic compounds are a kind of refractory organic matter whose unique chemical structures consist of an aromatic ring with one or more hydroxyl functional groups. Phenolic compounds used to produce pharmaceuticals, insecticides, cosmetics and plastics are discharged into the environment from chemical plants, oil refineries, paper and pulp mills (Wang et al. 2012). Many researchers have confirmed the presence of phenolic compounds in surface waters (Bolz et al. 2001; Gao et al. 2008; Song et al. 2015; Zhou et al. 2017; Zhong et al. 2018). The toxicity of phenols not only hinders survival and reproduction of aquatic organisms, but also endangers human health (Wang et al. 2012; Wolff et al. 2015). In particular, some phenolic compounds are known to be endocrine-disrupting compounds (EDCs), which have a detrimental effect on the endocrine system, such as 2,4,6-trichlorophenol (2,4,6-TCP). In 2011, the World Health Organization (WHO) listed 2,4,6-TCP and pentachlorophenol (PCP) as 2B pollutants with carcinogenic effects. Several phenolic compounds have been listed as priority pollutants by the US Environmental Protection Agency and European Union (EU) (Brinda et al. 2015). And six phenolic compounds are also listed as priority pollutants in water in China (Zhou et al. 1991).

Perennial rivers are major sources of freshwater for aquatic ecosystems worldwide. The phenolic compounds in rivers have been extensively researched in the world (Kawahata et al. 2004; Michalowicz & Duda 2007). Many scholars in China also studied these pollutants from different rivers. These studied rivers included the Pearl River (Diao et al. 2017), the Yellow River (Gao et al. 2008), the Yangtze River (Gao et al. 2008), rivers in Tianjin City (Zhong et al. 2018), Huai River (Gao et al. 2008), Songhua River (Wang et al. 2012) and Liaohe River (Li et al. 2015), which were all located in East, Northeast or South China. Northwest China, with an area of 3.2 million square kilometers, is an arid and semi-arid region. It belongs to a relatively undeveloped area. There are also many rivers located in this area. However, phenolic pollution in rivers over such a large area has never been studied.

Ecological risk assessment is a systematic approach that describes, explains and organizes scientific facts, laws and relationships, so as to provide a sound basis to develop adequate protective measures for the environment. It is also an essential tool for environmental management and is widely applied in environmental decision-making (Nash et al. 2008). Some scholars studied the ecological risk of phenolic compounds in surface waters from different regions of China (Jin et al. 2011; Zhong et al. 2018).

The Yellow River is the fifth largest river in the world, and is very important in China. Weihe River is the largest tributary of the Yellow River, with a drainage area of 134,800 square kilometers (Wang et al. 2016, 2019a). The river is called the ‘Mother River’ of the Guanzhong region, and plays an important role in the development of Northwest China and the health of the ecosystem of the Yellow River (Wei et al. 2012). It mainly flows through Baoji, Xianyang, Xi’an and Weinan cities in the Guanzhong region, and then flows into the Yellow River. In addition, Weihe River Basin also plays an important role in the Silk Road Economic Belt. Xi’an City in this basin was one of
four ancient capitals of the world and the starting point of the ancient Silk Road. It has a population of more than 10 million now. With the rapid increase in population and economic wealth, the insufficient water supply and degradation of the water environment has become one of the most restrictive factors for Weihe River basin’s economic development. Previous relevant studies on organic pollutants in Weihe River water mainly focused on polycyclic aromatic hydrocarbons (PAHs) (Chen et al. 2015) and organochlorine compounds (Wang et al. 2016), which had an adverse effect on aquatic organisms and human health. This is the first systematic study on the distribution and ecological risk of phenolic compounds in a river in Northwest China.

The aims of the present work were to identify and quantify 11 phenolic compounds in water, SPM and sediment samples and to assess the ecological risk associated with phenolic compounds present in Weihe River. The results not only contribute to a better understanding of the status of phenolic compounds pollution in rivers in arid, semi-arid and underdeveloped regions of the world, but also provide data for the development of countermeasures and management programmes for polluted rivers.

MATERIALS AND METHODS

Chemicals and instruments

Mixture standard solutions containing 11 kinds of phenolic compounds including phenol, 2-chlorophenol (2-CP), 4-chloro-m-cresol (4-Cmc), 2,4-dichlorophenol (2,4-DCP), 2,4,6-TCP, PCP, 4-nitrophenol (4-NP), 3-methylphenol (3-MP), 2,4-dinitrophenol (2,4-DNP), 2-nitrophenol (2-NP), 2,4-dimethyl phenol (2,4-DMP) were purchased from Putian Tongchuang Biotechnology Co. Ltd (Beijing, China). All chemical solvents (methanol, dichloromethane, glacial acetic acid, ethyl acetate and hydrochloric acid etc.) used for sample processing and analysis were of high-performance liquid chromatography (HPLC) grade or equivalent.

The phenolic compounds were analyzed using an ultra-high performance liquid chromatography (Waters H-class) coupled with UV detector (Waters Corp., USA). Chromatographic separation and resolution was achieved by using an ACQUITY UPLC® HSS T3 column (2.1 mm × 150 mm, 1.8 μm particle size, Waters Corp., USA). The mobile phase consisted of water with 0.5% formic acid (A) and acetonitrile with 0.5% acetic acid (B). The flow rate of the mobile phase was kept at 0.2 mL/min and the wavelength of determination was 280 nm. The gradient program was as follows: 0 min, 20% B; 10 min, 60% B; 15 min, 80% B.

Sampling area and sample collection

The Weihe River is located in northwest China, and supplies the drinking water and irrigation water for the cities along the river. There has been a dramatic increase in the emissions of industrial wastewater and urban domestic sewage in this area in recent years (Song et al. 2018). The Weihe River has many tributaries. The Jinghe River is the largest tributary of Weihe River, which has annual sediment runoff of approximately 9.46×10⁶ t. The large tributaries including Qianhe River, Jinling River, Jinghe River and Beiluo River have high sediment concentrations. There are many industrial enterprises in the basin, such as machinery, electronics, coal, and chemical industry. In Chencang District of Baoji section in the upper reaches of Weihe River, livestock farms are more common in the nearby Dazhangsi village. There are sewage treatment plants and sewage outlets along the middle reaches of Weihe River. The lower reaches of Weihe River is a traditional farming area. Chemical fertilizer is frequently used and the wastewater from the surrounding chemical fertilizer plants is directly discharged without treatment. Land use in the Guanzhong section of the Weihe River Basin is mainly dominated by cropland, forestland and grassland.

Based on the natural environmental conditions and the overall layout of the urban planning, a total of 32 sampling sites along Weihe River and its tributaries were selected for this sample collection (Figure 1). Nine, twelve and eleven sites were located in the upper (sites 1–9), middle (sites 10–21) and lower (sites 22–32) reaches of Weihe River, respectively. All samples were collected on August 30–31, 2017. On 31 August, the Weihe river flow at site 30 was 215 m³/s. During the whole sampling process global position system (GPS) was used to locate the sampling stations. In each sampling site, water samples were taken from 50 cm below the surface level with 4 L pre-cleaned brown glass bottles. All water samples were transported to the laboratory directly after sampling and kept at 4°C before analysis.

The SPM samples are derived from water samples in the laboratory. Water samples were filtered through a previously kiln-fired (400°C overnight) GF/F glass fibre filter (47 mm × 0.7 μm; Whatman, Maidstone, UK). Filters (suspended particulate matter, SPM) were kept in the dark at −20°C until analysis. Dissolved phase refers to the fraction of contaminants passing through the filter. This includes the
compounds that are both truly dissolved as well as those associated with colloidal organic matter. These filtrates were kept in the dark at 4 °C and extracted within 24 h.

Surface sediment (0–20 cm) samples were collected by using a grab sampler (Van VeenBodemhappe 2 L, Kiel, Germany) and put into aluminium containers. The sediments were transported to the laboratory and kept at −20 °C before analysis.

**Phenolic compounds extraction**

**Water**

A solid phase extraction (SPE) cartridges system from Supelco (Sigma Aldrich Corp., Saint Louis, MO, USA) was used for the enrichment of phenolic compounds in water. A volume of 1 L of water sample was filtered through a GF/F glass fiber filter in the laboratory. Next, 3 g of NaCl was dissolved in water, and samples were acidified with hydrochloric acid to pH 5. SPE columns containing 500 mg of styrene-divinyl benzene copolymer (Poly-Sery PSD) were activated with 6 mL of ethyl acetate, 6 mL of methanol and 6 mL of ultra-pure water successively. Then 1 L of water sample was percolated through the cartridges with a flow rate of 5 mL/min under a vacuum pump. The phenolic compounds were eluted to a glass tube by 5 mL ethyl acetate. The solvent fractions were then evaporated on a rotary evaporator at 35 °C, and exchanged by acetonitrile to a final volume of 1 mL.

**SPM**

SPM content was determined by gravimetry, after drying the glass fibre filter in an air-heated oven (55 °C until constant weight) and equilibrated at 25 °C in a desiccator. SPM containing glass fibre filter was cut into pieces, and then extracted three times by ultrasonic-assisted solvent extraction in 20 mL of dichloromethane for 1 h followed by centrifugation at 5,000 r/min for 10 min. Then 10 mL of supernatant was filtered through a silica gel column (3 g) with 10 mL (5 mL each time) elution of dichloromethane. The solvent fractions were then evaporated at 35 °C on a rotary evaporator, and exchanged by acetonitrile with a final volume of 0.5 mL and transferred into a vial through an organic phase filter membrane (0.22 μm) for subsequent testing.

**Sediment**

Sediments were oven dried at 40 °C. Then the dry sediment samples were carefully collected, homogenized and passed through a 250 μm standard sieve. Phenolic compounds in 3 g sediment were extracted three times by ultrasonic-assisted solvent extraction with 20 mL of dichloromethane for 1 hour followed by centrifugation at 5,000 r/min for 10 min. Then 10 mL of supernatant was filtered through a silica gel column (3 g) with 10 mL (5 mL each time) elution of dichloromethane. The solvent fractions were then evaporated at 35 °C on a rotary evaporator, and exchanged by
acetonitrile with a final volume of 1 mL and transferred into a vial through an organic phase filter membrane (0.22 μm) for subsequent testing. In this experiment, 3 parallel samples were made for each sample and 3 parallel blanks were made for each batch of samples.

**Parameters determination**

Physiochemical parameters such as chemical oxygen demand (COD), electrical conductivity (EC), total dissolved solids (TDS), salinity and pH value in water samples were measured immediately after arrival at the laboratory. The pretreatment process of the collected water sample was completed in 24 hours. The total organic carbon (TOC) of sediment was determined by TOC analyzer (TOC-VCPH; Shimadzu Corp., Shimadzu, Japan). The COD of water samples were determined by COD determinator (Shanghai, China). The EC, TDS and salinity were determined by a HQ30D water quality analyzer (Shanghai, China). The pH value was determined by a pH meter (Mettler Toledo, Columbus, OH, USA).

**Quality assurance/control (QA/QC)**

After every 5 samples, a procedural blank and a spike sample consisting of all reagents was run to check for interference and cross contamination, and no interferences or cross contamination were found in the whole test process. The limit of detection (LOD) and limit of quantification (LOQ) were calculated as having signal-to-noise ratios of above 3 and 10, respectively, by seven replicate analyses. The LODs of 11 phenolic compounds ranged from 0.005 to 0.02 μg/L in water and from 0.006 to 0.02 μg/g in sediment samples, respectively. The LOQs were in the range of 0.02–0.05 μg/L in water and 0.02–0.5 μg/g in sediment samples, respectively. The matrix spike recoveries for 11 phenolic compounds in water, SPM and sediment samples were determined by COD determinator (Shanghai, China). The RQ value was determined by a pH meter (Mettler Toledo, Columbus, OH, USA).

**Risk assessment method of phenolic compounds**

Ecological risk assessment of phenolic compounds was based on the US EPA ecological risk assessment framework, according to the method by Chakraborty et al. (2016) and Chen et al. (2014). The risk quotient (RQ) method was used to estimate the ecological risk of 11 phenolic compounds from Weihe River water. The RQ value was estimated as follows:

\[
RQ = \frac{MEC}{PNEC}
\]

where RQ was the risk characterization coefficient or the quotient value; PNEC was the predicted no effect concentration of compounds (μg/L); MEC was the measured environmental concentration of the compounds in water samples (μg/L).

The PNEC values for 11 phenolic compounds were obtained by using a species sensitivity distribution (SSD) approach. It was mainly performed in four steps: toxicity data screening and collecting for the pollutants, distribution model selection and SSD curve fitting, HC5 and PNEC calculation, and estimation of the ecological risk. First, the acute toxicity data (LC50 or EC50) and chronic toxicity data (NOEC, LOEC, EC10 and MATC) used in the current study for ecological risk assessment of 11 phenolic compounds in Weihe River water were obtained from the Ecotoxicology Database of the US EPA (https://cfpub.EPA.Gov/ecotox/[OL]). These selected datas involved three aquatic organisms (algae, invertebrates and vertebrates). According to principles for screening toxicological data for organic pollutants, when the experimental environment is fresh water such as lakes or rivers, the minimum exposure time for acute toxicological data is 96 h for vertebrates, 48 h for invertebrates and 24 h for algae, and the minimum exposure time for chronic toxicological data is 14 d for vertebrates, 7 d for invertebrates and 3 d for algae. Finally, the chronic toxicity datas of 5 phenolic compounds (phenol, 2,4-DNP, 2,4-DCP, 2,4,6-TCP and PCP) and the acute toxicity datas of the remaining 6 phenolic compounds (4-NP, 3-MP, 2-CP, 2-NP, 2,4-DMP and 4-CmC) were screened and used for the SSD model (Tables S1 and 2). Then, the SSD model curves were constructed using the National Institute of Public Health and the Environment, Bilthoven, Netherlands, based on a normal distribution (log-normal) software ETX 2.1. The HC5 (hazardous concentration for 5% of species) value was calculated based on the SSD curve and the PNEC value was obtained from the ratio between the HC5 and an assessment factor (AF).

The ecological risk levels were classified into three levels according to the RQ values. RQ > 1 indicates a high ecological risk, RQ values between 0.1 < RQ < 1 mean a median risk, and RQ < 0.1 suggests a minimal environmental risk (Wang et al. 2019a, 2019b).
RESULTS AND DISCUSSION

Phenolic compounds in water

The concentration ranges and mean values of 11 phenolic compounds in water samples collected from the Weihe River are summarized in Table 1. The concentrations of $\Sigma PC_{11}$ ranged from 0.06 to 14.12 $\mu$g/L with a mean value of 5.22 $\mu$g/L. The concentration range of $\Sigma CP_3$ was 0.05–11.6 $\mu$g/L with an average concentration of 4.19 $\mu$g/L, which accounted for 80.12% of the concentration of $\Sigma PC_{11}$. And the concentrations of $\Sigma NCP_6$ ranged from nd to 10.53 $\mu$g/L with a mean value of 1.04 $\mu$g/L, accounting for 19.88% of the total amount. The contents of phenol and chlorophenols were relatively higher than those of other phenolic compounds. The mean concentration of PCP was the highest (2.65 $\mu$g/L), accounting for more than half of the concentration of $\Sigma CP_3$. The detection frequencies of five phenolic compounds including 4-NP, 2,4-DNP, 2,4-DMP, 4-CmC, 2,4,6-TCP and PCP exceeded 50%, and the detection frequencies of PCP even reached 100%.

The data obtained in this study were compared with the data in published researches. It is difficult to directly compare our data with literature data due to the differences in compounds, sample number, sampling location and sampling season considered in these studies. However, the comparison can still provide useful information to understand the pollution levels of phenolic compounds in Weihe River.

Table 1 | Concentration ranges and mean values of phenolic compounds in water, SPM and sediment samples from Weihe River

| Phenolic compounds | Water (μg/L) | | | Sediment (μg/g) | | | |
|--------------------|-------------|---|---|----------------|---|---|
|                    | Range       | Mean ± SD | Range       | Mean ± SD | Range       | Mean ± SD |
| phenol             | nd-3.84     | 0.66 ± 1.11 | 0.02–1,715  | 101 ± 122 | 0.03–1.99   | 0.24 ± 0.38 |
| 4-NP               | nd-1.15     | 0.11 ± 0.23 | 0.01–34.1   | 3.45 ± 4.22 | 0.001–8.28 | 0.57 ± 1.53 |
| 3-MP               | nd-1.74     | 0.07 ± 0.31 | 0.06–13,449 | 954 ± 548 | 0.04–3.47   | 1.81 ± 0.85 |
| 2,4-DNP            | nd-2.69     | 0.12 ± 0.47 | 0.01–97.3   | 13.0 ± 24.0 | 0.004–1.24 | 0.13 ± 0.29 |
| 2-NP               | nd-0.38     | 0.03 ± 0.08 | 0.01–74.9   | 9.83 ± 4.29 | 0.003–0.08 | 0.04 ± 0.02 |
| 2,4-DMP            | nd-0.84     | 0.05 ± 0.15 | 0.03–121    | 13.6 ± 24.8 | 0.01–0.35   | 0.07 ± 0.06 |
| 2-CP               | nd-1.74     | 0.24 ± 0.45 | 0.02–940    | 66.1 ± 142 | 0.01–2.86   | 0.17 ± 0.54 |
| 4-CmC              | nd-4.53     | 0.54 ± 1.01 | 0.03–441    | 36.9 ± 21.0 | 0.01–1.52   | 0.19 ± 0.28 |
| 2,4-DCP            | nd-3.76     | 0.39 ± 0.88 | 0.12–1,639  | 155 ± 47.5 | 0.01–15.58  | 2.15 ± 4.28 |
| 2,4,6-TCP          | nd-2.48     | 0.35 ± 0.56 | 0.16–2,028  | 194 ± 101 | 0.03–22.72  | 3.05 ± 4.71 |
| PCP                | 0.03–5.96   | 2.65 ± 1.29 | 0.37–22,021 | 2,902 ± 1,245 | 0.43–7.79 | 2.66 ± 2.21 |
| $\Sigma NCP_6$     | nd-10.53    | 1.04 ± 1.99 | 0.15–13,993 | 1,091 ± 425 | 0.80–12.32  | 2.87 ± 2.01 |
| $\Sigma CP_3$      | 0.03–11.6   | 4.19 ± 2.77 | 0.76–20,892 | 3,555 ± 1,416 | 0.76–52.21  | 8.22 ± 8.43 |
| $\Sigma PC_{11}$   | 0.06–14.12  | 5.22 ± 3.46 | 0.92–34,885 | 4446 ± 1,424 | 3.54–34.09  | 11.09 ± 8.68 |

The total concentrations of 2,4-DNP, 2,4,6-TCP and PCP in water from the Weihe River (0.03–12.20 μg/L) were higher than those in Yangtze River (0.0036–1.004 μg/L) (Gao et al. 2008). The maximum of total concentrations of 2,4,6-TCP and PCP in water from the Weihe River (8.44 μg/L) were much higher than those in Huaihe River (421 ng/L) and Haihe River (110 ng/L) (Gao et al. 2008). However, compared with rivers such as the Beitang Drainage River (nd–45.1 μg/L) and Dagu Drainage River (nd–106.1 μg/L) (Zhong et al. 2018), the total concentrations of phenol and 2,4-DMP in Weihe River (nd–4.68 μg/L) were lower. The total concentrations of 2,4-DNP, 2,4,6-TCP and PCP in Weihe River were also lower than those in the Yellow River basin (48.68 μg/L) (Gao et al. 2008). Weihe River in this study is a tributary of the Yellow River. However, the research area for Gao et al. (2008) was located in the main stream of the Yellow River. It may be due to this reason that the sources of phenolic compounds in Weihe River and main stream of the Yellow River are different. The concentration of phenol in Weihe River (nd–3.84 μg/L) was much lower than that in Messinian Rivers (0.02–1.108 mg/L) (Anastasopoulou et al. 2015). The above comparisons suggested that the concentration of phenolic compounds in water from Weihe River was at a moderate level.

Compared with other phenolic compounds, the concentrations of 2,4-DCP, 2,4,6-TCP and PCP were higher in Weihe River water. However, none of them exceeded the limit values in Standard Limits for Specific Projects of...
Surface Water Sources of Centralized Drinking Water (SLSPSWCDW) (GB 3858 2002), as shown in Table 2.

**Phenolic compounds in SPM**

As shown in Table 1, the concentrations of $\sum$PC$_{11}$ in SPM samples ranged from 0.92 to 34,885 $\mu$g/g with a mean value of 4,446 $\mu$g/g. The concentrations of $\sum$NCP$_{6}$ ranged from 0.15 to 13,993 $\mu$g/g with a mean value of 1,091 $\mu$g/g, accounting for 24.5% of the concentrations of $\sum$PC$_{11}$. The concentration range of $\sum$CP$_{5}$ was 0.76–20,892 $\mu$g/g, accounting for 75.5% of the concentrations of $\sum$PC$_{11}$. The detection frequencies of all the 11 phenolic compounds reached 100%, while in the water samples, only the detection frequency of PCP reached 100%. For individual phenolic compounds, the average concentration of PCP was also the highest (2,902 $\mu$g/g), accounting for 65.3% of the total amount.

**Phenolic compounds in sediment**

The results of phenolic compounds in Weihe River sediments were shown in Table 1. The concentrations of $\sum$PC$_{11}$ in SPM accounted for 65.3% of the total amount. The concentration range of $\sum$PC$_{11}$ was 0.76–32.21 $\mu$g/g with an average concentration of 8.22 $\mu$g/g, accounting for 74.15% of the total amount. The concentration range of $\sum$NCP$_{6}$ was 0.80–12.32 $\mu$g/g with an average concentration of 2.87 $\mu$g/g, accounting for 25.85% of $\sum$PC$_{11}$. The concentration range of 2,4,6-TCP was from 0.03 to 22.72 $\mu$g/g with a mean concentration of 3.05 $\mu$g/g; which was the highest among the 11 phenolic compounds. The concentration of PCP was the second highest (range: 0.43–7.79 $\mu$g/g; mean ± SD: 2.66 ± 2.21 $\mu$g/g). The reason why the content of chlorophenol compounds was much higher than that of non-chlorophenol may be that chlorophenol compounds were widely used as raw materials such as for disinfectants, organic pesticides, resins and surfactants in the production of paper, printing, anti-corrosion, dyes, pharmaceuticals and other industries. Industrial wastewater has been discharged into rivers in more and more ways, resulting in a higher concentration in Weihe River sediments.

The concentration of phenolic compounds in sediment from Weihe River was lower than that in Liaohe River basin (Li et al. 2015). However, the total concentration ranges of phenol, 2-CP and 2,4-DMP in Weihe River sediments were 0.05–5.20 $\mu$g/g, which were much higher than those in Beitang Drainage River (nd–16.79 $\mu$g/kg) (Zhong et al. 2018). The total concentration of phenol and PCP in Weihe River sediments (0.46–9.78 $\mu$g/g) were much higher than those in Daqiao Drainage River (nd–71.07 $\mu$g/kg) (Zhong et al. 2018). The concentration range of $\sum$CP$_{5}$ (0.76–32.21 $\mu$g/g) in Weihe River sediments was much higher than that in Lake Mariut, Egypt (24.9–1,246 $\mu$g/kg) (Khairy 2015). These comparisons indicated that the concentrations of phenolic compounds in Weihe River sediments were at a relatively high level. This may be due to the fact that the water body had been greatly influenced by industries such as the dyeing and chemical industries for a long time, which caused the pollutants in the water to settle into the river sediment, thus resulting in the higher concentration of phenolic compounds in sediments of Weihe River.

**The spatial distribution of phenolic compounds**

The spatial distribution of phenolic compounds in water, SPM and sediments were studied by comparing the concentrations of 11 phenolic compounds among upper (sites 1–9), middle (sites 10–21) and lower (sites 22–32) reaches (Figure 2). It could be observed that both in water and sediment, the highest concentrations of $\sum$PC$_{11}$ were observed in the middle reach, while the lowest concentrations of $\sum$PC$_{11}$ were found in the lower reach. This may be due to the sampling time being August, which belongs to the wet season of the river. Moreover, the largest tributary of Weihe River, Jinghe River and Weihe River, intersects between site 22 and site 23, which increased the runoff of the lower reach. Therefore, the content of phenolic compounds in the water of the lower reach was relatively lower than that of the middle and upper reaches. The higher concentrations of $\sum$PC$_{11}$ in water were observed in site 12, Xingping Sewage Plant (site 15), Luohe River (site 23) and site 24 (Figure 3(a)). These sampling sites were all located near sewage outlets. Large quantities of waste water contained phenolic compounds were discharged into the middle reach of Weihe River, which caused the $\sum$PC$_{11}$ in water to be highest in the middle reach. The

| Phenolic compounds | Limit value | Maximum value in Weihe River water |
|-------------------|-------------|-----------------------------------|
| 2,4-DCP           | 93          | 3.76                              |
| 2,4,6-TCP         | 200         | 2.48                              |
| PCP               | 9           | 5.96                              |
highest concentrations of $\Sigma_{PC_{11}}$ were observed in site 15, site 16 and site 19 in sediment (Figure 3(c)). These sites were all located in the Xianyang or Xi’an areas, which were the most developed areas in this basin with some sewage treatment plants nearby. The results may be caused by the discharge of urban domestic sewage and industrial wastewater. Also, the water in the middle reach was flat and the SPM adsorbed phenolic compounds from water, then sank to the bottom of the river, which caused the $\Sigma_{PC_{11}}$ in sediment to increase. For SPM, however, the highest concentrations were found in the upper reach and the lowest concentrations were found in the lower reach. The SPM content in the lower reach was the highest (Figure 3(b)).

For individual phenolic compounds, the mean concentration of PCP in water was the highest in each of the three reaches (Figure 2(a)). In the upper and lower reaches, the phenol concentration was the second highest, while in the middle reaches 2,4-DCP concentration was the second highest. For SPM samples, the mean concentration of PCP was the highest too (Figure 2(b)). Also, the 3-MP concentration was the second highest in the upper and middle reaches, differing from that in water samples. The concentration of each phenolic compound was lowest in the lower reaches. For sediment samples, the highest concentrations of individual phenolic compounds were 3-MP in upper reaches, 2,4-DCP in middle reaches and PCP in lower reaches, respectively.
The percentages of $\Sigma NCP_6$ and $\Sigma CP_5$ in water, and SPM and sediment in different reaches, were also calculated (Figure 4). In any reach of Weihe River, the percentage of $\Sigma CP_5$ was higher than that of $\Sigma NCP_6$. So it can be concluded that the chlorophenol pollution in the Weihe River was more serious than non-chlorophenol pollution. The highest percentages of $\Sigma CP_5$ were found in the middle reaches in every kind of sample.

The distribution of phenolic compounds between water, SPM and sediment

It is believed that the environmental fate and behavior of hydrophobic organic compounds is ultimately determined
by the physicochemical properties of each compound and sediment, such as organic content, size distribution, partition coefficient and salinity (Chen et al. 2015). Physicochemical parameters such as COD (7.14–1.025 mg/L), EC (512.1–1.565 us/cm), TDS (214–1.432 mg/L), salinity (0.23–0.62 ppt) and pH (7.46–9.84) in water samples from Weihe River were detected, but no significant correlations were observed between concentrations of phenolic compounds and these physiochemical parameters.

Many researchers hold that the TOC value determined the concentrations of hydrophobic organic compounds in sediment and positive linear relations could be found between hydrophobic organic compound concentrations and the TOC values in sediment (Witt 1995; Kannan et al. 2001; Chen et al. 2015). The TOC of sediments in Weihe River ranged from 0.91 to 8.06% (Table 3) with a mean value of 2.1%. But no linear relationship was found between phenolic compounds concentrations and TOC (Figure S1). This may be due to the following reasons. First, the phenolic compounds and TOC come from different sources. Second, the hydrophobicities of some phenolic compounds are weak. Although sediments and water in river systems such as Weihe River undergo dynamic sorption and desorption, the concentrations of hydrophobic organic compounds in SPM and these physiochemical parameters.

| Site No. | $\sum_{PM11}$ in Water (g/L) | $C_{PM11}$ (g/L) | $K_d$ (L/g) | TOC (g/g) | $K_d$ (L/g) |
|---------|-----------------------------|-----------------|-------------|---------|-------------|
| 1       | 5.59                        | 0.013           | 6.635       | 1.18    | 2.53        | 6.20        | 1.1 |
| 2       | 1.54                        | 0.018           | 4.536       | 2.951   | 2.73        | 3.54        | 2.3 |
| 3       | 8.74                        | 0.024           | 7.903       | 904     | 1.91        | 6.42        | 0.7 |
| 4       | 8.25                        | 0.002           | 51.956      | 3.829   | 1.75        | 3.54        | 0.4 |
| 5       | 9.17                        | 0.010           | 22.167      | 2.417   | 1.29        | 12.38       | 1.4 |
| 6       | 2.60                        | 0.024           | 9.579       | 3.679   | 1.62        | 4.76        | 1.8 |
| 7       | 5.06                        | 0.006           | 5.228       | 1.032   | 1.55        | 5.28        | 1.0 |
| 8       | 8.14                        | 0.052           | 871         | 107     | 1.51        | 4.29        | 0.5 |
| 9       | 11.9                        | 0.077           | 1.743       | 146     | 2.07        | 12.43       | 1.0 |
| 10      | 11.5                        | 0.007           | 25.638      | 2.269   | 1.26        | 16.96       | 1.5 |
| 11      | 7.17                        | 0.234           | 498         | 69      | 2.39        | 3.96        | 0.6 |
| 12      | 13.9                        | 0.064           | 2.624       | 189     | 2.24        | 19.30       | 1.4 |
| 13      | 16.8                        | 0.043           | 3.616       | 215     | 2.13        | 4.58        | 0.3 |
| 14      | 4.61                        | 0.013           | 6.453       | 1.399   | 0.91        | 3.58        | 0.8 |
| 15      | 1.86                        | 0.659           | 241         | 130     | 1.61        | 27.33       | 14.7 |
| 16      | 12.0                        | 1.06            | 171         | 14      | 1.79        | 22.11       | 1.8 |
| 17      | 4.00                        | 0.080           | 722         | 180     | 3.10        | 6.77        | 1.7 |
| 18      | 3.88                        | 0.027           | 7.956       | 2.051   | 1.28        | 34.09       | 8.8 |
| 19      | 3.65                        | 0.106           | 918         | 252     | 2.01        | 10.12       | 2.8 |
| 20      | 0.80                        | 0.078           | 2.785       | 3.461   | 2.11        | 8.80        | 10.9 |
| 21      | 13.0                        | 0.058           | 1.980       | 152     | 1.54        | 9.99        | 0.8 |
| 22      | 7.83                        | 0.022           | 7.989       | 1.020   | 1.73        | 6.57        | 0.8 |
| 23      | 21.2                        | 0.116           | 1.046       | 49      | 1.38        | 21.49       | 1.0 |
| 24      | 33.1                        | 0.126           | 3.590       | 108     | 1.48        | 11.12       | 0.3 |
| 25      | 4.71                        | 0.023           | 3.107       | 660     | 8.06        | 6.41        | 1.4 |
| 26      | 11.0                        | 7.58            | 7.8         | 0.71    | 1.60        | 5.64        | 0.5 |
| 27      | 3.83                        | 0.214           | 581         | 152     | 3.27        | 6.13        | 1.6 |
| 28      | 2.63                        | 2.23            | 26          | 10      | 1.38        | 5.57        | 2.1 |
| 29      | 6.46                        | 7.42            | 2.6         | 0.41    | 1.43        | 8.15        | 1.3 |
| 30      | 6.37                        | 17.9            | 4.0         | 0.63    | 2.07        | 8.56        | 1.3 |
| 31      | 5.09                        | 22.7            | 0.9         | 0.18    | 2.59        | 6.42        | 1.3 |
| 32      | 3.23                        | 5.21            | 16          | 4.9     | 1.98        | 5.04        | 1.6 |

(2018); as a result, the distribution coefficient tends to be closely related to the properties of contaminants, in particular their octanol-water partition coefficient ($K_{ow}$). The organic carbon normalized partition coefficient of phenols (i.e. $K_{oc}=K_d/f_{oc}$) in Weihe River were plotted against $K_{ow}$ values (Figure 5). Some reports agreed that log$K_{oc}$ increased with log$K_{ow}$ of organic compounds, consistent with the
so-called linear free energy relationship (Zhou & Maskaoui 2003; Zhu et al. 2008). However, no significant linear correlation between log\(K_{oc}\) and log\(K_{ow}\) was found in Weihe River. As shown in Figure 5, the slope, intercept and square determination coefficient were 0.66, 1.87 and 0.48, respectively. The significant difference between \textit{in situ} \(K_{oc}\) is perhaps related to the disequilibrium between the concentrations of sediment samples and surface water samples. Many reasons including dilution, high surface runoff and atmospheric fallout could produce the disequilibrium (Zhong et al. 2018).

### Ecological risk assessment of phenolic compounds in Weihe River

The specific risk characterization results were shown in Table 4. Among the 11 phenolic compounds, 8 phenolic compounds exhibited minimal (RQ < 0.1) or median (RQ: 0.1–1) ecological risks in all sampling sites. For these 8 phenolic compounds, the sampling sites with RQ < 0.1 accounted for more than 80%, with 0.1 < RQ < 1 accounted for a small proportion (below 20%) and with RQ > 1 accounted for 0%, that was, these 8 phenolic compounds did not pose high ecological risks to aquatic organisms in the study area. However, RQ values of 4-CmC, 2,4-DCP and PCP in some sites were more than 1.0 (meaning a high ecological risk). The proportions of high risk of 4-CmC and 2,4-DCP were 53.13% and 21.88%, respectively. The RQ value of PCP ranged from 0.034 to 6.83 and the rate of high risk was the highest among 11 phenolic compounds, reaching 93.75%, which indicated that the Weihe River suffered serious ecological risk from PCP. Based on the above analysis, 4-CmC, 2,4-DCP and PCP should be selected as priority phenolic compounds for Weihe River and some effective measures should be taken to control and reduce the pollution of these phenolic compounds.

### CONCLUSIONS

In this study, we conducted single sampling due to insufficient funding and some other reasons. However, analyses of 11 phenolic compounds still provided very useful information for the evaluation of the contamination levels, spatial distribution and ecological risks that are located in Northwest China. The results obtained in this study showed that the levels of phenolic compounds in Weihe River were at a moderate level for water and at a relatively high level for sediment in the world, respectively. In any reach of Weihe River, the concentration of \(\sum CP5\) was higher than that of \(\sum NCP6\). For individual phenolic compounds, PCP pollution was the most serious. The results of ecological risk assessment of phenolic compounds by RQs indicated a high ecological risk in Weihe River water. 4-CmC, 2,4-DCP and PCP had higher contribution to the total phenols in the study area. Thus, more attention should be paid to these phenolic compounds, and

![Figure 5](http://iwaponline.com/wst/article-pdf/83/8/2012/880503/wst083082012.pdf)
preventing greater risks to the environment. In most sites, the distribution coefficient ($K_d$) (SPM) was much higher than $K_d$ (sediment), which probably suggested fresh phenolic compounds input in Weihe River. The existence of these fresh phenolic compounds and their potential risk should also be given more rigorous evaluation.

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DATA AVAILABILITY STATEMENT

All relevant data are included in the paper or its Supplementary Information.

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