Source of gold in Neoarchean orogenic-type deposits in the North Atlantic Craton, Greenland: Insights for a proto-source of gold in sub-seafloor hydrothermal arsenopyrite in the Mesoarchean

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1. Introduction

On Earth, the abundance of a given element (e.g., gold, Au), for example in the continental crust, may be either (1) the result of its partitioning from the internal material as a result of the differentiatiation of Earth’s bulk composition and reorganization of its interior structure through the impact of core-mantle-plate geodynamics (Shahar et al., 2019), or, (2) the result of a preserved extra-terrestrial input supplied over a prolonged period of time (e.g., iridium enrichment in the cap carbonate sequences – an outcome from the demise of the Cryogenian Snowball Earth; Bodiselitsch et al., 2005). In fact, unlike osmium and iridium, which were sequestered into the core in the incipient Earth between 4.51 and > 3.90 Ga and replenished by the late veneer at ca. 3.90 to 3.80 Ga (Fig. 1A & B; Brenan and McDonough, 2009; Willbold...
et al., 2011; Barboni et al., 2017), gold mostly remained in the material that would differentiate into the young Earth’s mantle (Fig. 1B; Brenan and McDonough, 2009). Therefore, it is critical to understand how and when a “proto-source” of gold was extracted from the Earth mantle and made available into the mafic crust precursor through first-stage melting of the mantle in the Archean. Second-stage melting of this mafic precursor, which produced more felsic rocks, would re-distribute gold in a more mature crust comprising tonalite-trondhjemite-granodiorite (TTG) bodies and granite-greenstone belts (Shirey and Richardson, 2011; Næraa et al., 2012; Tang et al., 2016; Smit et al., 2019; O’Neil et al., 2019; Johnson et al., 2019; Laurent et al., 2020).

Previous benchmark studies have suggested that an overwhelming proportion of the gold in present-day Earth’s continental crust may have been concentrated at ca. 2.9–2.7 Ga at the time of the unique interplay between atmospheric, hydrodynamical and biological conditions occurring to the formation of palaeoplacers in the Mesoarchean (i.e., the “Mesoarchean gold event”; Frimmel, 2014, 2018; Heinrich, 2015; Fig. 1C). Then, to explain the distribution of gold deposits in the Earth crust through geological times (Groves et al., 2005; Goldfarb et al., 2001, 2010; Frimmel, 2018; O’Neil et al., 2019; Johnson et al., 2019; Laurent et al., 2020).

Building on previous research for the Paleoarchean, we explore the concept envisaging that a “pool” of gold existed in the geothermally active mafic crust of the Mesoarchean Earth (Large et al., 2015; Hofmann et al., 2017). Thus, we first present a review on the pool of gold during the Paleoarchean (Large et al., 2015; Hofmann et al., 2017) and the origin of Mesoarchean palaeoplacers (Frimmel, 2014, 2018; Heinrich, 2015) to piece together what is currently known about the endowment of the crust in gold from the Paleoarchean until the Mesoarchean gold event.

In contrast to the low fertility of Paleoarchean granite-greenstone belts for lode gold deposits (also referred as “orogenic gold deposits”) and given a sharp decrease in the abundance of komatiites within the Earth’s crust from ca. 3.2 Ga (Greber et al., 2017; Fig. 1C), we postulate that the Mesoarchean mafic crust could have retained its gold and been a fertile source for Neoarchean structurally-controlled lode gold deposits in Mesoarchean greenstone belts. Examples of such structurally-controlled gold deposits of Neoarchean age, which are coeval with amphibolite-facies metamorphism of mafic protoliths in greenstone belts (e.g., ca. 2.64 Ga gold-bearing arsenopyrite mineralisation at Storø; Scherstén et al., 2012), exist in the Greenland fragment of the North Atlantic Craton (GNAC; inset in Fig. 2A).

In the Tartoq District within the GNAC (Fig. 2A), a complex patchwork of mafic crustal rocks and protoliths of serpentiniﬁed harzburgite formed at ca. < 3.19–3.01 Ga (Szilas et al., 2013, 2014; van Hinsberg et al., 2018; Polat et al., 2016; van Hinsberg et al., 2018). Several shear zones affect those rocks and host gold mineralisation (Appel and Secher, 1984; Petersen, 1991; Kisters et al., 2012; Kolb et al., 2013), in particular in the Nuuluk area in the Tartoq greenstone belt (Fig. 2A and B), where gold is associated with massive arsenopyrite (FeAsS) and pyrite (FeS2). Here, we contribute new evidence about the nature of crust-forming processes in...
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2. Current understanding of the endowment of the crust in gold in the
Paleoarchean and Mesoarchean

The bimodal felsic and mafic crust in Archean cratons may be the
result of (1) a first-stage melting of the early mantle producing a mafic
crustal precursor, and, (2) a second-stage melting of this mafic pre-
cursor yielding felsic rocks (O’Neil et al., 2019). In his model for the
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belts in the hinterland of the Archean cratons. Instead a source of the
gold in palaeoplacers is explained by the unique interplay of (i) atmos-
pheric conditions, (ii) ocean and riverine water chemistry, (iii) atmos-
pheric weathering driven by pH2(g) causing non-redox acidic dissolu-
don of Fe(II)-species or reduction of Fe(III)-species (Hao et al., 2019),
and (iv) evolution of life at ca. 2.9 Ga that triggered continent-wide
weathering, transfer of gold to reduced seawater as Au(HS)2− and, sub-
sequent large-scale trapping of gold in and around coastal environments
in the Mesoarchean (Heinrich, 2015; Large et al., 2015; Frimmel,
2019). This source of gold in seawater (Fig. 1C; Large et al., 2015;
Hofmann et al., 2017) was readily explained by the extensive transfer of
gold to the anoxic Paleoarchean ocean, with limited incorporation of
gold into carbonaceous chert, as a result of low-temperature (< 150 °C)
hydrothermal seafloor alteration and pervasive silification of an

3. Geology of the Tartoq greenstone belt

The GNAC is fringed by Paleoproterozoic orogenic belts to the north
and south (Kolb et al., 2013 and references therein). Several > 3600 Ma
to ca. 2550 Ma tectonic terranes or blocks of Archean rocks were in-
volved in a major accretionary orogeny at ca. 2720–2700 Ma (Nutman
and Friend, 2007; Kolb et al., 2012; Dziongel et al., 2014, 2017),
with cratonisation concluding at ca. 2560 Ma (Friend et al., 1988, 1996;
Nutman and Friend, 2007; Nutman et al., 1989, 2004). At the south-
western margin of the GNAC (inset in Fig. 2A), the greenstone belt
comprising the Tartoq Group is found in six individual km-scale
synclinal structures that all host several orogenic gold occurrences
(Fig. 2A; Nielsen, 1976; Appel and Secher, 1984; Kolb et al., 2013;
Steenfelt et al., 2016). The Mesoarchean Tartoq Group comprises mafic
rocks with well-preserved primary volcanic and magmatic features
(e.g., subaqueous lava flows including evidence of pillow lavas, vol-
canic breccias) together with serpentinites of harzburgite cumulate
protooliths that originally formed by hydrous melting of a shallow
mantle source (Szilas et al., 2013, 2014). These rocks are interpreted as
remnant of a < 3190–3012 Ma Mesoarchean oceanic crust involved in
an accretionary complex (Kisters et al., 2012; Szilas et al., 2013, 2014;
Polat et al., 2016; van Hinsberg et al., 2018; Fig. 2A). Accumulation and
stacking of this hydrated oceanic crust resulted in the formation of ca.
3012–2993 Ma felsic mylonites in thrusts, and imbrication of the Tartoq
Group rocks with voluminous TTGs as young as 2824 Ma (Nutman
and Friend, 2004; Nutman et al., 2014). These TTGs may have formed by
the melting of thickened oceanic crust (Nagel et al., 2012; Hastie et al.,
2016) in a “repeatedly-aborted subduction zone” setting (Kisters et al.,
2012; Szilas et al., 2013, 2014; van Hinsberg et al., 2018). As a con-
sequence of this tectonic evolution, in the Nuuluk part of the Tartoq
greenstone belt, which is the focus of the present study, the Tartoq
Group including lenses of ultramafic rocks was originally metamor-
phosed to the amphibolite facies (~580 °C, ~4.5 kbar) prior to and

The Tartoq greenstone belt and gold mineralisation in the Greenland fragment of the North Atlantic Craton (GNAC). (A) Geological and structural map of the Tartoq district after Petersen (1991). Inset: Location of the GNAC and the main gold occurrences in southwestern Greenland after Kolb et al. (2013), (B) Geological map of the Nuuluk area in the Tartoq greenstone belt showing location of the gold-mineralised Western and Eastern Carbonate Zones (WCZ & ECZ, after Petersen (1991) and Szilas et al. (2013, 2014)). The photographs show the three samples selected for Re-Os-S isotope geochemistry studies. The whole-rock gold assays are
 compiled by Kolb (2011). Abbreviation: sup. alt.: supergene alteration.
possibly when imbricated into TTGs (Evans and King, 1993; Kisters et al., 2012; Szilas et al., 2014; van Hinsberg et al., 2018, van Hinsberg, pers. comm. 2019). Shear zones are characterised by localized greenschist-facies retrograde metamorphism that overprint the peak amphibolite-facies assemblage in the Nuuluk area (van Hinsberg et al., 2018). The shear zones (Fig. 2B) comprise an anastomosing network of auriferous quartz-ankerite ± pyrite veins, which either have an ankerite-chlorite-quartz-pyrite-pyrrhotite-chaalcopyrite-tennantite halo, or are locally walled by fuchsite ± tourmaline ± talc at the contact with serpentinite lenses (Evans and King, 1993; Kolb et al., 2013; van Hinsberg et al., 2018). Tourmaline thermometry and the retrograde assemblage of chloritoid-albite-epidote-titanite-calcite in the mafic rocks attest to conditions of 380 ± 50 °C at a pressure < 2 kbar (~4 km depth; van Hinsberg et al., 2018).

Two 25-m-wide and ca. 5-km-long shear zones (i.e., the gold-rich Eastern and gold-poorer Western Carbonate Zones – ECZ & WCZ, Fig. 2B), which lie 500 m apart and are located ca. 100–200 m into the footwall of a major thrust zone (King, 1985; Evans and King, 1993; Kolb, 2011; Kolb et al., 2013; Steenfelt et al., 2016), host gold mineralisation found in three assemblages: (type 1) as 5–20 µm anhedral grains in arsenopyrite and pyrite (or in fractures in those minerals) in massive sulphide layers, in particular in the gold-rich ECZ (Evans and King, 1993; up to 20 ppm Au); (type 2) in discordant, pinch-and-swell massive sulphide layers, in particular in the gold-rich ECZ (Evans and King, 1993; Kolb, 2011; Kolb et al., 2013; Steenfelt et al., 2016), host gold mineralisation found in three assemblages: (type 1) as 5–20 µm anhedral grains in arsenopyrite and pyrite (or in fractures in those minerals) in massive sulphide layers, in particular in the gold-rich ECZ (Evans and King, 1993; up to 20 ppm Au); (type 2) in discordant, pinch-and-swell massive sulphide layers, in particular in the gold-rich ECZ (Evans and King, 1993; Kolb, 2011; Kolb et al., 2013). The host rocks in the immediate vicinity of mineralisation comprise ankerite, chlorite, quartz and fuchsite, whereas carbonate, chlorite, tourmaline, and pyrite dominate away from the mineralised zones. The arsenopyrite-pyrite layers were interpreted as submarine exhalative in origin ("proto-ore") whereas the origin of chalcopyrite and tennantite was invoked as being related to an alteration of the primary massive sulphide layers during metamorphism (Appel and Secher, 1984).

4. Methods

4.1. Production of mineral separates of individual sulphide/sulpharsenide species

The protocol utilised in the present study to obtain mineral separates of individual sulphide species is described in detail in a companion method article entitled “Mineral separation protocol for accurate and precise rhenium-osmium (Re-Os) geochronology and sulphur isotope composition of individual sulphide species” and published in MethodX.

4.2. Re-Os isotope geochemistry

Following sulphide petrography and mineral separation, Re-Os isotope geochemistry of arsenopyrite and pyrite were carried out on samples TTQ-01, TTQ-02 and TTQ-05. For each analysis, between 188 and 530 mg of sulphide (arsenopyrite or pyrite) mineral separate were weighed and transferred into a thick-walled borosilicate Carius tube (Shirey and Walker, 1995). Each sulphide aliquot was dissolved in inverse Aqua Regia (~3 mL of 11 N HCl and ~6 mL 16 N HNO3) with a known amount of 185Re + 190Os spike” solution or 185Re + common Os spike” solution at 210 °C for 24 h (Laboratory for Sulphide and Source Rock Geochemistry and Geochronology in the Durham Geochemistry Centre, Durham University, UK; Selby et al., 2009). The 185Re + common Os spike” solution was used for arsenopyrite mineral separates identified to being bereft of common Os following analysis using the 185Re + 190Os spike” solution (Table 1). The Re-Os laboratory protocol used in the present work is described in full in Selby et al. (2009). The Re and Os isotopic compositions were determined by negative thermal ionization mass spectrometry (N-TIMS) using a ThermoScientific Triton mass spectrometer at the Arthur Holmes Laboratory in the Durham Geochemistry Centre, Durham University, UK. Rheon was measured as ReO4− in static mode on Faraday collectors, whereas Os was measured as OsO4− in peak-hopping mode on a SEM (Creaser et al., 1991; Völkering et al., 1991). Sulphide measurement quality was monitored by repeated measurements of i-house Re (125 pg aliquot – 185Re/187Re = 0.59892 ± 0.00203, n = 74) and Os (DROs – 50 pg aliquot, 187Os/188Os = 0.160869 ± 0.000410, n = 100) standard solutions. Total procedural blanks for each set of samples are reported in Table 1. The analytical uncertainties result from full error propagation of weighing errors, spike calibration, standard measurements, mass spectrometry analyses and blanks. When using the 185Re + common Os spike” solution, the Re-Os ages are determined through regression of Re-Os data in 187Os/188Os vs. 187Re/188Os space using 2σ level absolute uncertainties and the error correlation, rho, using Isoplot v 4.15 (Ludwig, 2011) with the 187Re decay constant of Smoliar et al. (1996; λ187Re = 1.666e−11 ± 1.165e −14 a−1). Those Re-Os ages are based on the computation of Model 1 isochrons. Alternatively, the other Re-Os ages are determined as model ages (t) for aliquots analysed using the 185Re + common Os spike” solution by using the formula: t = ln ([187Os*/187Re] + 1)/λ187Re), where 187Os* is the measured content of radiogenic 187Os.

After preliminary Re tests to estimate the Re content in each sample, full Re-Os isotope geochemistry isotope procedures were carried out using appropriate volumes of the 185Re + 190Os” spike solution. For arsenopyrite aliquots of sample TTQ-02, considering the blank levels that we report (Table 1), no less than 84–89% of 188Os (i.e., the normalizing isotope of the common Os fraction) was contributed by the blank. At those blank levels, for a range of assumed initial 187Os/188Os ratios of 0.11, 1, 5 or 10, the Os budget in those aliquots comprise 99.97–99.98% radiogenic 187Os. Therefore, considering an Os budget comprising 100% radiogenic 187Os, aliquots of arsenopyrite from sample TTQ-02 were spiked using the 185Re + common Os spike”. As such, individual more precise model ages could be determined for each aliquot in the way described above.

4.3. Sulphur isotopic composition of the arsenopyrite and pyrite mineral separates

The sulphur isotopic composition of the sulphide and sulpharsenide mineral species are combined with the Re-Os ages and the initial Os isotopic composition of those minerals, where available, to constrain the source(s) of Os and S. Approximately 5–10 mg of arsenopyrite or pyrite mineral separate was utilised for each isotopic analysis. Sulphides were analysed by standard techniques (Robinson and Kusakabe, 1975) at the Scottish Universities Environmental Research Centre – SUERC, Glasgow, UK. The liberated gases were analysed on a VG Isotech SIRA II mass spectrometer, and standard corrections applied to raw δ34S values to produce true δ34S. Repeat analyses of international and SUERC standards NBS-123, IAEA-S-3, and CP-1 gave δ34S values of +17.1‰, −32% and −4.6‰, respectively, with a standard error of ±0.3% during the execution of these samples. Data are reported in δ34S notation as per mil (%) variations from the Vienna Cañon Diablo Troilite (V-CDT) standard.

5. Results

5.1. Sulphide petrography and paragenetic sequence

The paragenetic sequence, which is valid for the massive sulphide bodies in both the ECZ and the WCZ, is presented in Fig. 3A. In the ECZ, euchedral to mostly subhedral, medium- to coarse-grained (< 4 mm) arsenopyrite forms a textural layering with subhedral to anhedral,
Table 1
Synopsis of the Re-Os-S isotope geochemistry data for arsenopyrite and pyrite at Nuuluk, Tartoq Greenstone Belt.

### Preliminary full Re-Os isotope geochemistry procedure using the $^{185}\text{Re} + ^{190}\text{Os}$ spike solution

| Sample ID | Tectono-stratigraphic position | Mineral | Sample weight (mg) | Re ± 2σ | Total Os ± 2σ | $^{187}\text{Re}$ ± 2σ | $^{192}\text{Os}$ ± 2σ | $^{187}\text{Os}^*$ ± 2σ | % 187Os* | $^{187}\text{Re}/^{188}\text{Os}$ ± 2σ | % Re blank | % 187Os blk | % 188Os blk |
|-----------|-------------------------------|---------|-------------------|---------|----------------|----------------|----------------|----------------|-----------|--------------------|-----------|-----------|-----------|
| TTQ-APY-02 | WCZ | Arsenopyrite | 393.50 | 1.648 | 0.006 | 45 | 206 | 1.036 | 0.004 | 0 | 45 | 49.97 | 509043 | 1649046 | 22138 | 71722 | 0.35 | 0.007 | 88.97 |
| TTQ-APY-02 | WCZ | Arsenopyrite | 396.06 | 1.732 | 0.006 | 48 | 244 | 1.088 | 0.004 | 0 | 48 | 49.98 | 598631 | 2151666 | 26378 | 94820 | 0.34 | 0.004 | 84.33 |

### Full Re-Os isotope geochemistry procedure using the $^{185}\text{Re} + \text{common Os}$ spike solution

| Sample ID | Tectono-stratigraphic position | Mineral | Sample weight (mg) | Re ± 2σ | Total Os ± 2σ | $^{187}\text{Re}$ ± 2σ | $^{187}\text{Os}^*$ ± 2σ | Model age ± 2σ | % Re blank | % 187Os blk | % 188Os blk |
|-----------|-------------------------------|---------|-------------------|---------|----------------|----------------|----------------|----------------|-----------|-----------|-----------|
| TTQ-02-01-APY/01 | WCZ | Arsenopyrite | 411.90 | 1.509 | | 0.949 | 0.009 | 51.7 | 0.4 | 3184 | 42 | 1.3% |
| TTQ-02-01-APY/02 | WCZ | Arsenopyrite | 520.99 | 1.528 | | 0.960 | 0.007 | 53.1 | 0.3 | 3316 | 33 | 1.1% |
| TTQ-02-02-APY/01 | WCZ | Arsenopyrite | 498.01 | 2.115 | | 1.330 | 0.054 | 67.4 | 0.3 | 2967 | 124 | 4.2% |
| TTQ-02-02-APY/02 | WCZ | Arsenopyrite | 582.92 | 2.204 | | 1.385 | 0.056 | 68.3 | 0.3 | 288 | 120 | 4.2% |
| TTQ-02-02-APY/03 | WCZ | Arsenopyrite | 290.82 | 2.263 | | 1.422 | 0.015 | 73.1 | 0.8 | 3099 | 47 | 1.6% |
| TTQ-02-02-APY/04 | WCZ | Arsenopyrite | 510.27 | 2.588 | | 1.627 | 0.065 | 78.3 | 0.4 | 2822 | 118 | 4.2% |

### Full Re-Os isotope geochemistry procedure using the $^{185}\text{Re} + ^{190}\text{Os}$ spike solution

| Sample ID | Tectono-stratigraphic position | Mineral | Sample weight (mg) | Re ± 2σ | Total Os ± 2σ | $^{187}\text{Re}$ ± 2σ | $^{187}\text{Os}^*$ ± 2σ | $^{187}\text{Re}/^{188}\text{Os}$ ± 2σ | % Re blank | % 187Os blk | % 188Os blk |
|-----------|-------------------------------|---------|-------------------|---------|----------------|----------------|----------------|--------------------|-----------|-----------|-----------|
| TTQ-01-APY/01 | ECZ | Arsenopyrite | 416.39 | 430 | 2 | 114 | 2 | 42 | 20 | 1 | 1.0 | 0.1 | 0.699 | 1.28 |
| TTQ-01-APY/01 | ECZ | Arsenopyrite | 510.18 | 385 | 4 | 109 | 2 | 40 | 2 | 19 | 1 | 1.1 | 0.1 | 0.685 | 1.22 |
| TTQ-01-APY/03 | ECZ | Arsenopyrite | 188.96 | 534 | 11 | 119 | 4 | 43 | 25 | 2 | 1.3 | 0.1 | 0.688 | 2.38 |
| TTQ-01-APY/04 | ECZ | Arsenopyrite | 372.80 | 456 | 5 | 548 | 10 | 221 | 9 | 4 | 0.2 | 0.3 | 0.677 | 1.41 |
| TTQ-05-PY/01 | ECZ | Pyrite | 331.15 | 585 | 2 | 70 | 3 | 22 | 53 | 4 | 2.4 | 0.3 | 0.707 | 1.19 |
| TTQ-05-PY/02 | ECZ | Pyrite | 242.60 | 369 | 2 | 78 | 4 | 28 | 22 | 3 | 1.4 | 0.2 | 0.707 | 2.57 |
| TTQ-05-PY/03 | ECZ | Pyrite | 363.87 | 311 | 1 | 834 | 29 | 341 | 28 | 1.8 | 0.1 | 0.0 | 0.706 | 2.01 |
| TTQ-05-PY/04 | ECZ | Pyrite | 430.36 | 333 | 5 | 25 | 1 | 6.2 | 4 | 106 | 7 | 5.1 | 0.4 | 0.703 | 1.68 |
| TTQ-05-PY/05 | ECZ | Pyrite | 370.89 | 364 | 5 | 19 | 1 | 3.5 | 0.3 | 208 | 19 | 9.5 | 1.1 | 0.730 | 1.78 |
| TTQ-05-PY/06 | ECZ | Pyrite | 461.04 | 340 | 4 | 18 | 1 | 3.4 | 0.2 | 200 | 12 | 9.0 | 0.7 | 0.734 | 1.53 |
| TTQ-02-PY/02 | WCZ | Pyrite | 425.95 | 960 | 3 | 45 | 1 | 5.1 | 0.1 | 376 | 5 | 20.8 | 0.3 | 0.701 | 0.56 |
| TTQ-02-PY/03 | WCZ | Pyrite | 555.71 | 347 | 4 | 27 | 3 | 5.6 | 0.1 | 123 | 13 | 7.5 | 1.6 | 0.493 | 1.24 |
| TTQ-02-PY/04 | WCZ | Pyrite | 521.77 | 351 | 4 | 16 | 1 | 1.5 | 0.1 | 480 | 34 | 26.9 | 2.4 | 0.790 | 1.31 |
| TTQ-02-PY/04 | WCZ | Pyrite | 566.47 | 346 | 4 | 35 | 4 | 7.1 | 0.9 | 97 | 12 | 8.2 | 1.8 | 0.570 | 1.37 |
| TTQ-02-PY/05 | WCZ | Pyrite | 442.41 | 275 | 4 | 20 | 2 | 2.1 | 0.1 | 158 | 17 | 10.9 | 2.2 | 0.506 | 1.65 |

### Blank type Os ± 2σ | Re ± 2σ

| Sample ID | Tectono-stratigraphic position | Mineral | % 187Os * | % 188Os blk |
|-----------|-------------------------------|---------|-----------|-----------|
| TTQ-01-APY/01 | ECZ | | | |
| TTQ-01-APY/01 | ECZ | | | |
| TTQ-01-APY/03 | ECZ | | | |
| TTQ-01-APY/04 | ECZ | | | |
| TTQ-05-PY/01 | ECZ | | | |
| TTQ-05-PY/02 | ECZ | | | |
| TTQ-05-PY/03 | ECZ | | | |
| TTQ-05-PY/04 | ECZ | | | |
| TTQ-05-PY/05 | ECZ | | | |
| TTQ-05-PY/06 | ECZ | | | |
| TTQ-02-PY/02 | WCZ | | | |
| TTQ-02-PY/03 | WCZ | | | |
| TTQ-02-PY/04 | WCZ | | | |
| TTQ-02-PY/04 | WCZ | | | |

187Os*: radiogenic 187Os.
(*) uncertainty on the decay constant of 187Re is included.
coarse-grained (0.5 mm– < 4 mm) pyrite (TTQ-01, Fig. 3B). In the WCZ (Fig. 3C), euhedral to subhedral coarse-grained (ca. 1–2 mm) arsenopyrite occurs between layers of younger medium- to coarse-grained subhedral to anhedral pyrite (ca. 0.25– > 2 mm; TTQ-02-sub-sample-01 in Fig. 2B). Locally, euhedral to subhedral arsenopyrite, which is associated with quartz and younger very coarse-grained pyrite (TTQ-02-sub-sample-02 in Fig. 2B), is very coarse-grained (> 2 mm).

Arsenopyrite exhibits localized features of weak recrystallization and intense brecciation prior to cementation by quartz (Fig. 3D). Arsenopyrite, which can be found as rounded or irregular inclusions in pyrite, was locally dissolved and partially replaced by pyrite (Fig. 3B, C, E). Locally, replacement of arsenopyrite by pyrite has resulted in the healing/destruction of fractures and brittle features within the arsenopyrite. Rare chalcopyrite and tennantite occur with pyrite. In the vicinity of lenses of serpentinites in the ECZ, open-space filling by fuchsite resulted in a pinch-and-swell pattern of pyrite-quartz veins (Fig. 3F, G).

5.2. Re-Os-S isotope geochemistry of monophasic mineral separates

In the WCZ, arsenopyrite ($^{34}$S = +1.8 ± 0.2‰ V-CDT, 2σ) is bereft of common Os and only contains radiogenic $^{187}$Os ($^{187}$Os$^*$; Table 1). Two aliquots of the same mineral separate of euhedral to subhedral coarse-grained (ca. 1–2 mm) arsenopyrite in between layers of younger medium- to coarse-grained subhedral to anhedral pyrite (ca. 0.25– > 2.00 mm, sub-sample TTQ-02-01, “sub-01” in Fig. 2B) yield reproducible and precise Re-Os data (1.51–2.59 ppb Re and 51.5–78.3 ppt $^{187}$Os$^*$) with overlapping Mesoarchean Re-Os model ages of 3136 ± 33 Ma and 3184 ± 42 Ma (Fig. 4A; 1.1–1.3%, 2σ, with the uncertainty including the $^{187}$Re decay constant uncertainty; Smoliar et al., 1996). Four aliquots of the same mineral separate of euhedral to subhedral very coarse-grained (> 2 mm) arsenopyrite associated with quartz and younger very coarse-grained pyrite (sub-sample TTQ-02-02, “sub-02” in Fig. 2B) yield slightly less precise Re-Os data (2.12–2.59 ppb Re and 67.4–78.3 ppt $^{187}$Os$^*$) which yield overlapping Re-Os model ages of 2967 ± 124 Ma, 2889 ± 120 Ma, 3009 ± 47 Ma, 2822 ± 118 Ma and a weighted average of 2972 ± 118 Ma and a weighted average of 2972 ± 120 Ma (4.0%, 2σ, Fig. 4A; n = 4, 95% confidence, mean square weighted deviates, MSWD = 3.6). Pyrite (n = 5, sample TTQ-02), which contains common Os, possesses lower Re and Os concentrations (0.275–0.960 ppb Re and 16–45 ppt total Os) than arsenopyrite in this sample. The regression of the Re-Os data of five pyrite aliquots from this sample, with a limited range in $^{187}$Re/$^{188}$Os values (4–34), but positively correlated with the significantly radiogenic $^{187}$Os/$^{188}$Os compositions (7.5–26.9), yields an imprecise Mesoarchean–Neoarchean Re-Os isochron age of 2860 ± 108 Ma (Fig. 4C; 2σ, Model 1 isochron; MSWD = 2.2, initial $^{187}$Os/$^{188}$Os (Os i) ratio = 0.13 ± 0.02).

Unlike arsenopyrite in the WCZ, arsenopyrite from the ECZ (sample TTQ-01; $^{34}$S = +2.1 ± 0.2‰) has lower Re concentrations (385–534 ppt) and contains common Os (for which a best approximation is given by the $^{192}$Os contents of 30–221 ppt, ca. 35–40% of the total Os contents of 84–548 ppt). The regression of the Re-Os data of the four arsenopyrite aliquots, with a limited range in $^{187}$Re/$^{188}$Os values (4–28), but positively correlated with the low to moderately radiogenic $^{187}$Os/$^{188}$Os compositions (0.3–1.4), yielded a Neoarchean Re-Os isochron age of 2608 ± 108 Ma (Fig. 4C; 2σ, Model 1 isochron; MSWD = 3.0, initial $^{187}$Os/$^{188}$Os (Os i) ratio = 0.13 ± 0.02).
Similarly, pyrite (sample TTQ-05, δ34S = +1.8 ± 0.2‰) also possesses low Re and Os concentrations: 311–585 ppt Re and 18–834 ppt total Os (192Os = 3.4–341 ppt). Collectively, the Re-Os data of pyrite (n = 6, sample TTQ-05) yield, within uncertainty, an identical Neoarchean Re-Os isochron age of 2656 ± 89 Ma (2σ, Model 1 isochron, MSWD = 0.71, Osi ratio = 0.12 ± 0.02; Fig. 4D).

6. Discussion

6.1. Hydrothermal alteration of oceanic crust and sub-seafloor gold-bearing arsenopyrite precursor

Arsenopyrite was the first mineral phase to precipitate in the form of massive mineralised bodies (Fig. 3). Arsenopyrite subsequently underwent minor recrystallization, brittle deformation, and finally, local dissolution and replacement by pyrite (Fig. 3). In the WCZ, arsenopyrite did not incorporate any common Os into its structure upon precipitation. As such, its present Os budget only comprises radiogenic 187Os produced by the isobaric decay of 187Re. The weighted average of four individual Re-Os model ages constrained in the present study for arsenopyrite (n = 4, sub-sample 02, TTQ-02) in the WCZ. The individual Re-Os model ages for sub-sample 01, TTQ-02 (n = 2) yield a maximum age for massive arsenopyrite in the WCZ; B. Model 1 Re-Os isochron diagram in the 187Re/188Os vs. 187Os/188Os space for pyrite aliquots (n = 5) in the WCZ; C. Model 1 Re-Os isochron diagram in the 187Re/188Os vs. 187Os/188Os space for arsenopyrite aliquots (n = 4) in the ECZ; D. Model 1 Re-Os isochron diagram in the 187Re/188Os vs. 187Os/188Os space for pyrite aliquots (n = 6) in the ECZ. Ellipses are constructed from the maximum and minimum error vectors that are orthogonal to one another. Maximum and minimum uncertainties are statistical values that are calculated from the uncertainty of the 187Os/188Os and 187Re/188Os ratios for a given data point. Final uncertainties were calculated by full error propagation of uncertainties in the Re and Os measurements, blank values, isotopic compositions, spike calibrations, and reproducibility of the standard Re and Os values. The error correlation function, rho, is utilised for isochron regressions. The uncertainty in the 187Re decay constant is included in both the isochron and model age uncertainty (after Smoliar et al. (1996) and Ludwig (2011)).
massive sulphides rather than stringer-type footwall mineralisation in volcanogenic massive sulphide deposits in the Phanerozoic (Heinrich and Eadington, 1986; Lydon, 1988; Hannington et al., 1999; Brueckner et al., 2015).

At Tartoq, vestiges of harzburgitic protoliths occur in the form of serpentinite. The process of serpentinisation may have caused Re and As depletion (present-day As contents of 0.6–28 ppm) from an ultramafic protolith that was produced by hydrous melting of the Mesoarchean mantle (Szilas et al., 2014). Arsenic, which is the most soluble chalcophile element, can be leached by hydrothermal fluids from ultramafic rocks that typically contain ca. 10–450 ppm As (Smedley and Kinniburgh, 1990; Hannington et al., 1999; Patten et al., 2015, 2019). Given the fact that As controls the accumulation of Au in Fe-sulphides and Fe-sulpharsenides (Deditius et al., 2014; Xing et al., 2019), and that As remains in solution until an As-dominant mineral phase (e.g., arsenopyrite) precipitates (Zhong et al., 2015), precipitation of hydrothermal arsenopyrite could have sequestered significant proportions of gold. Elevated Au contents in black smoker fluids and alteration profiles of present-day oceanic crust are compatible with gold being leached during hydrothermal alteration of oceanic crust (Falkner and Edmond, 1990; Hannington et al., 1999; Patten et al., 2015), and/or more importantly, in particular in island arc magmatic systems, high gold contents are explained by magmatic volatile exsolution together with S and Se (Hedenquist et al., 1993; Simmons and Browne, 2000; Rae et al., 2001; Simmons and Brown, 2006, 2007; Hannington et al., 2016; Patten et al., 2019). In the current geodynamic setting of Iceland setting which is proposed as an analogue for the formation of Earth’s earliest evolved crust (Reimink et al., 2014), sub-seafloor sulphide mineralisation may result from the boiling of chloride-bearing and neutral fluids (3.2 wt% NaCl, pH = 5–6) before discharging at black smoker seafloor vents (Hardardóttir et al., 2009, 2010; Hannington et al., 2016). Considering this analogue, we propose that such a hydrothermal sulphide mineralisation, which is highly enriched in Au and As in present-day systems (Au up to 590 ppm; Hardardóttir et al., 2009; Patten et al., 2016, 2019; Fuchs et al., 2019), could have taken the form of sub-seafloor gold-bearing arsenopyrite bodies during hydrothermal alteration of oceanic crust in the Mesoarchean.

The chemical stability of arsenopyrite in near-surface and hydrothermal environments was re-evaluated as being higher than previously thought (Pokrovski et al., 2002). Indeed, even in the presence of pyrite, arsenopyrite remains chemically stable providing that it is kept in a water-saturated and moderately reduced environment at a pH above 5. At those conditions, water in equilibrium with arsenopyrite should have dissolved arsenic concentrations in the 0.01–0.10 ppm range (Craw et al., 2003). Furthermore, the Mesoarchean shallow-marine “oxygen oases” (Eickmann et al., 2018; Ossa Ossa et al., 2019), in which reduced aqueous iron species were oxidized, would have contributed to strip the ocean from dissolved Fe\(^{2+}\) and triggered a local change from ferruginous to sulphidic waters in shallow to middle level ocean domains (Large et al., 2015). In an anoxic environment with less than 2% O\(_2\), sulphide-driven mobilization of arsenic from arsenopyrite does not occur and arsenopyrite is not dissolved (Zhu et al., 2008). Thus, collectively taken, these thermodynamic properties and the neutral to slightly acidic pH of seawater in the Mesoarchean ocean (ca. 6.5–7.0; Shibuya et al., 2018; Haley and Bachan, 2017; Krissansen-Totton et al., 2018) support our model and explain the stability of massive arsenopyrite bodies in sub-seafloor settings in the Mesoarchean, even in the case where a section of the oceanic crust could have been opened to “oxidizing” conditions in a shallow part of the overall reduced Mesoarchean ocean.

We suggest that the sulphur isotopic composition (\(^{34}\)S = +1.8 ± 0.2‰) of the ca. 2972 Ma arsenopyrite, which was preserved through medium-grade metamorphism in the Nuuluk Greenstone belt (see Section 6.2), could fit with a magmatic source of reduced sulphur. Generally, sulphur leached from igneous wall rocks or derived from magmatic fluids can account for \(^{34}\)S values between ca. 0 and +5‰ for sulphides in volcanic-associated deposits (Huston, 1999). Yet, the identification of the source of sulphur must discriminate between all potential pools of sulphur, including: (1) isotopic fractionation in open vs. partly open systems; (2) sulphate reduction via bacterial pathways; (3) igneous rocks and magmatic fluids. In a geothermally active Mesoarchean oceanic crust (Polat et al., 2007), hydrothermal systems in the oceanic crust could be seen as open to seawater input. Seawater in the Mesoarchean was probably mostly ferruginous with higher sulphate contents (200 μM > sulphate ≥ 5 μM) only in shallow-marine environment where a mass-dependent fractionation of ca. 20‰ between biogenic sulphides (\(^{34}\)S = ca. −20‰) and sulphate (\(^{34}\)S = ca. +3 to +8‰) could occur (Farquhar et al., 2010; Large et al., 2015; Eickmann et al., 2018). The sulphur isotopic composition of hydrothermal arsenopyrite is not
compatible with this pathway of sulphate reduction. In contrast, in agreement with the premise by Huston (1999), although not ignoring the possibility of the contribution by Mesoarchean, locally sulphidic, middle-level seawater with $\delta^{34}S$ values around 0‰ (Farquhar et al., 2007, 2010), we favour the interpretation that Mesoarchean hydrothermal systems in seafloor/sub-seafloor setting could have brought reduced sulphur derived from the Tarqot Group magmatic rocks and serpentinitised harzburgite with a likely mantle-type sulphur isotopic composition traditionally considered to be 0 ± 2‰ (Thode et al., 1961; Seal, 2006), despite local sulphur heterogeneities in the early Earth mantle (Farquhar et al., 2002). Furthermore, a source of Au in the form of magmatic volatile degassing would have been accompanied by S (Patten et al., 2019) recording a magmatic sulphur isotopic signature.

In conclusion, a proto-source of Au in the Mesoarchean oceanic crust corresponds to sub-seafloor “proto-ores” made of hydrothermal arsenopyrite with primary invisible gold that formed in connection with basalt-hosted geothermal systems in an oceanic volcanic arc setting (Fig. 5). Similar systems comprising massive to semi-massive Au-Ag-Cu-rich sulphide showings have been recognized in Neoarchean greenstone belts of the Superior Province, Canada, where (1) syn-volcanic sulphides are associated with palaeo-hydrothermal alteration of basic volcanic lavas with pillows and breccias, and (2) massive to semi-massive arsenopyrite-pyrite are found in shear zones in those greenstone belts (Galloway et al., 2019).

Direct evidence from petrographic observations and absolute geochronology show that this “proto-ore” is present and preserved in the WCZ where it possibly underwent subsequent yet localized tectono-metamorphic recrystallisation during prograde metamorphism. In contrast, in the ECZ, the existence of such a proto-ore is based on petrographic evidence and comparison with the WCZ to explain the resetting of the Re-Os chronometer in the massive arsenopyrite Mesoarchean “proto-ore” and incorporation of common Os derived from local surrounding rocks in the ECZ through the action of retrograde tectono-metamorphic overprint.

6.2. Prograde and retrograde metamorphism of the massive arsenopyrite precursor and gold mineralisation though secondary enrichment at 2.66 Ga

The imbrication of the Tarqot Group rocks with TTG at ca. 3012–2828 Ma is interpreted as the result of the stacking of short lived and disrupted ‘slabs’ of hydrated oceanic crust in an accretionary complex (Fig. 5; Nutman et al., 2004; Kisters et al., 2012; Szilás et al., 2013, 2014; Polat et al., 2016). This geodynamic setting overlapped and post-dated > 3.0 Ga prograde metamorphism of the Nuuluk part of the Tarqot greenstone belt to the amphibolite facies (ca. 580 °C, 4.5 kbar; van Hinsberg et al., 2018). Brittle deformation and brecciation of the competent massive arsenopyrite layers could have been coeval with or have followed this event. Yet, arsenopyrite bearing primary invisible gold remained chemically robust during amphibolite-facies metamorphism (Fougereous et al., 2016a) and brittle deformation at Nuuluk.

Shear zones subsequently localized greenschist facies retrograde metamorphism that overprinted the peak amphibolite facies assemblage at Nuuluk (van Hinsberg et al., 2018; Fig. 5). During the retrograde metamorphism (380 ± 50 °C and < 2 kbar), small amounts of crystal plasticity of arsenopyrite is possible (e.g., with dissolution-reprecipitation of arsenopyrite), especially if the deformation vectors are at a high angle to the preferred orientation of arsenopyrite (Fougereous et al., 2016a). In the massive arsenopyrite bodies, our petrographic observations show that pyrite precipitated where arsenopyrite was locally dissolved (Fig. 3E). This local dissolution of arsenopyrite could be a consequence of its oxidation (i.e., As$^{3+}$ converted to As$^{5+}$) due to reduction of water at anodic sites on arsenopyrite crystal surfaces (Walker et al., 2006). In addition, during the strain-event arsenopyrite dissolution, the loss of gold from the crystal lattice is facilitated by localized domains of recrystallisation, most likely due to fluid percolation along sub- and new grain boundaries (Fougereous et al., 2016a). Therefore, in the ECZ where the gold abundances peak (20–109 ppm; Fig. 2B), the stoichiometric dissolution of a precursor arsenopyrite with primary invisible gold by small volumes of relatively low $f_{O_2}$, chlorine-bearing (~0.01 M HCl) solutions may have occurred as follows (Pokrovski et al., 2002; Fougereous et al., 2016a):

$$\text{FeAsS} + 3\text{H}_2\text{O} + 2\text{H}^+ + 2\text{Cl}^- \rightarrow \text{FeCl}_2(aq) + \text{As(OH)}_3(aq) + \text{H}_2$$

In fact, as exists predominantly as $[\text{As}^{+3}(\text{OH})_3]_{aq}$ (James-Smith et al., 2010; Kohl et al., 2017) in moderate temperature (> 200 °C, for this study area 380 ± 50 °C and < 2 kbar), CO$_2$-rich (0.05–0.25 mol %), S-bearing and low salinity (typically ≤3 wt% NaCl eq.) hydrothermal fluids responsible for “orogenic type” gold mineralisation (Mikucki, 1998; Phillips and Evans, 2004). In the ECZ, evidence for the presence of CO$_2$ in the fluids comes from the precipitation of ankerite in zones bearing gold mineralisation (Evans and King, 1993; Kolb, 2011; Kolb et al., 2013). The removal of CO$_2$ from the fluid through ankerite precipitation in the ECZ (Figs. 3A and 5) resulted in the activity of CO$_2$ not being high enough to continuously buffer the fluid pH. Therefore, suitable conditions for elevated gold concentration in the fluids as sulphide complexes of Au$^{+1}$, i.e., AuH$^+$ and Au(HS)$^-$ (Fougereous et al., 2016; Heinrich and Eadington, 1986; Phillips and Evans, 2004; Pokrovski et al., 2002) were no longer present, and gold sulphide complexes were destabilized and gold precipitation with accompanying pyrite occurred in two principal ways: (1) at the contact between arsenopyrite and pyrite over distances of a few micrometres in the massive arsenopyrite bodies (i.e., grain boundaries; in cracks of pre-existing arsenopyrite; or within newly formed pyrite; Pokrovski et al., 2014; Xing et al., 2019), or (2) over distances of several meters within quartz-pyrite veins that were later reopened during fuchsite formation (Fig. 3G; Appel and Secher, 1984; Evans and King, 1993). At ~380 °C (tourmaline thermometry; van Hinsberg et al., 2018), it is likely that significant sulphur isotope fractionation was hampered (Seal, 2006) between the timing of dissolution of the massive arsenopyrite precursor, which is capable of preserving its sulphur isotopic composition ($\delta^{34}S = +2.1‰$) through amphibolite-facies metamorphism (cf. Wagner et al., 2004), and subsequent pyrite precipitation as indicated by the overlapping sulphur isotopic compositions of arsenopyrite ($\delta^{34}S = +2.1‰$) and pyrite ($\delta^{34}S = +1.8‰$) in the ECZ.

In the ECZ, using petrographic and isotopic evidence, we propose that the loss of gold from the crystal lattice of the arsenopyrite protore through arsenopyrite dissolution under retrograde greenschist-facies conditions caused the resetting of the Re-Os isotopic system in arsenopyrite. Thus, the Re-Os isochron age of 2608 ± 108 Ma would record the best estimate for the timing of arsenopyrite dissolution. This arsenopyrite Re-Os isochron age overlaps with the pyrite Re-Os isochron age of 2656 ± 89 Ma. Considering the model of mineral precipitation presented above, the pyrite Re-Os age records the best estimate of the timing of free gold precipitation, i.e., gold that was originally present in the crystal lattice of the arsenopyrite precursor.

The significance of arsenopyrite as precursor for the formation of gold deposits is not a new concept but one that remains rather poorly explored (e.g., Fougereous et al., 2016b). Such a process of gold upgrading/secondary enrichment through metamorphism of an arsenopyrite “proto-ore” was also conceptualized to explain the origin of the Palaeoproterozoic Boliden Au-Cu-As massive sulphide deposit in Sweden (Wagner et al., 2004, 2007; Mercier-Langevin et al., 2013). The Boliden deposit records progressive recrystallization and porphyroidal growth of arsenopyrite (Wagner et al., 2007). It is proposed that invisible gold and sulphur were liberated from arsenopyrite, which underwent dissolution and replacement in response to metamorphism (Wagner et al., 2004, 2007). Both gold and sulphur were then precipitated in cross-cutting Au-vein rich in mineral phases that inherited the sulphur isotopic composition of the arsenopyrite precursor with invisible gold,
Os isotopic composition of the Mesoarchean mantle, preclude any ratios of Neoarchean arsenopyrite and pyrite, which overlap with the \( \frac{187\text{Re}}{188\text{Os}} = 0.435 \pm 0.055 \) and \( \frac{187\text{Os}}{188\text{Os}} = 0.130 \pm 0.001 \) for 3.20–2.80 Ga. Therefore, the overlapping and equivalent initial Os ratios of the Neoarchean arsenopyrite deposits, which were originally deposited in volcanic-arc setting, to produce younger lode gold deposits.

6.3. Insights into the palaeo-environmental conditions in arc setting in the Mesoarchean

In the WCZ, the Mesoarchean arsenopyrite proto-ore is shown to have been essentially bereft of common Os at the time of precipitation. Therefore, the overlapping and equivalent initial Os ratios of the Neoarchean arsenopyrite with reset Re-Os systematics (\( \text{Os}_{\text{apy-ECZ}} = 0.13 \pm 0.02 \)) and the Neoarchean neo-petraprecipitated pyrite (\( \text{Os}_{\text{py-ECZ}} = 0.12 \pm 0.02 \)) in the ECZ were not derived from Os present in a massive arsenopyrite precursor of presumed Mesoarchean age in the ECZ. Instead, it is most likely that arsenopyrite and pyrite acquired their initial Os isotopic composition from the surrounding serpentinsitised Mesoarchean harzburgite protoliths and volcanic rocks of the Tartoq Group. For these Mesoarchean Tartoq Group rocks, we suggest an original Os isotopic composition of 0.11 ± 0.01 that is equivalent to the hypothetical primitive Mesoarchean upper mantle (\( \text{Os}_{\text{PMUM}} \)) at 3200–2800 Ma, calculated by using the present-day values of \( \frac{187\text{Re}}{188\text{Os}} = 0.435 \pm 0.055 \) and \( \frac{187\text{Os}}{188\text{Os}} = 0.130 \pm 0.001 \) for a primitive upper mantle (Meisel et al., 2001; Carlson, 2005). The Os ratios of Neoarchean arsenopyrite and pyrite, which overlap with the \( \text{Os}_{\text{PMUM}} \) isotopic composition of the Mesoarchean mantle, preclude any addition of radiogenic Os that would have elevated the Os ratios of Neoarchean arsenopyrite and pyrite to values significantly higher than 0.11 at the time of mineral precipitation. Therefore, we dismiss a regional contribution in Os by crustal fluids released from evolved crustal lithologies (i.e., with high Re/Os ratios and significant accumulation of \( \text{Os} \) through decay of \( \text{Re} \)) involved in the accretionary complex. Instead, in the ECZ, Os is interpreted to have been only derived locally. In addition, the Os initial ratio of pyrite in the WCZ, for which petrographic evidence show that it post-dates and replaces arsenopyrite, seems to confirm this local derivation of Os by metamorphic fluids. Indeed, although imprecise, the Re-Os isochron age for pyrite in the WCZ possesses an Os ratio of 2.4 ± 1.3 (i.e, Os > 1.1). This Os ratio for pyrite in the WCZ is significantly higher than the estimate for the primitive Mesoarchean mantle.

The strength of our work lies in the geologically robust, high-quality Re-Os ages for arsenopyrite and pyrite. These new data, combined with our detailed petrographic observations and existing age data for the wider study area, support the conclusion that hydrothermal arsenopyrite related to arc volcanism formed in sub-seafloor setting in the Mesoarchean (arsenopyrite Re-Os model age), prior to acting as the principal source for lode gold in younger Neoarchean (pyrite Re-Os isochron age) orogenic-type deposits in the area.

In the present study, a model for a peculiar and non-negligible proto-source of gold in the Mesoarchean oceanic crust is emerging: (1) primary extraction of Au from the Mesoarchean mantle at the time of ca. < 3.19–3.01 Ga basalt magmatism and associated lower crustal harzburgite cumulates; (2) Au, Re, and Au contributed by magmatic volatile exsolution and alteration of the upper oceanic crust through hydrothermal cell circulation and serpentinisation of the lower crustal harzburgite cumulates; (3) hydrothermal systems precipitating sub-seafloor massive arsenopyrite bodies with invisible gold at ca. 3.18–3.13 Ga, (4) arsenopyrite retaining gold and remaining chemically robust during > 3.0 Ga prograde amphibolite-facies metamorphism and during imbrication of the Tartoq greenstone belt and TTGs at ca. 3.01–2.82 Ga; (5) retrograde greenschist-facies metamorphic overprint localized in shear zones at Nuuluk when the crystal plasticity of arsenopyrite caused its local dissolution, the resetting of its Re-Os geochronometer with a Neoarchean age in areas of major gold loss (in particular in the ECZ); (6) secondary enrichment of primary invisible gold in Mesoarchean massive arsenopyrite through retrograde metamorphism focused within shear zones into lodes in the juvenile continental crust that contain free gold associated with newly formed ca. 2.66 Ga pyrite.

A peak in lode gold or orogenic-type gold deposits occurred between ca. 2.75 and 2.55 Ga during a “Neoarchean Global Gold Event” (Goldfarb et al., 2001, 2005; Bierlein et al., 2006), which coincides with the time when the juvenile continental crust had become able to sustain plate tectonics and record strong regional deformation fabrics. Thus, in light of the two mineralising events identified in the Tartoq greenstone belt in the Greenland fragment of the North Atlantic Craton, we contribute a more general working hypothesis suggesting a connection between a ca. 3.18–3.13 Ga gold proto-source in sub-seafloor arsenopyrite in arc-related Mesoarchean greenstone belts, and, the ca. 2.75–2.65 Ga “Global Gold Event” representing the gold endowment of the juvenile continental crust in stabilising cratons through metamorphic upgrading of the Mesoarchean proto-source.

Competing interests

Within the last three years, both Joshua W. Hughes and Denis M. Schlatter have provided consultancy to AEX Gold Inc., the current exploration licence holder of the Tartoq greenstone belt. However, their involvement concerns the company’s other exploration properties and has not incorporated the Tartoq greenstone belt. The other authors declare no competing interest.

Acknowledgments

N. J. Saintilan (NJS) and D. Selby (DS) acknowledge the technical support by Dr. Geoff Nowell, Antonia Hofmann and Dr. Chris Otlely at Durham University. J. Kolb (JK) thanks the Geological Survey of Denmark and Greenland (GEUS) and the Ministry of Mineral Resources and Labour, Greenland for financially supporting field work and subsequent research in the area. Fieldwork in 2016 was carried out by...
Joshua W. Hughes (JWH) on behalf of Nanoq Resources Ltd., during gold exploration funded by the present license holder, AEX Gold Inc. Peter J. Dodds is thanked for assistance during the fieldwork. AEX Gold Inc. are acknowledged for shipment of the samples in this study. NJS thanks Dr. Nicolas Thebaud (University of Western Australia) and Dr. Patrick Mercier-Langevin (Geological Survey of Canada), and Prof. Dr. Hartwig Frimmel (University of Würzburg) for interesting discussions around an earlier version of the manuscript. We thank Editor Prof. Dr. Wilson Teixeira, the Associate Editor, and an anonymous reviewer for providing insightful comments and suggestions during the review process.

Funding
This work was supported financially through a Swiss National Science Foundation Advanced Postdoc.Mobility Grant (#P300P2_171496) awarded to NJS. DS acknowledges the TOTAL Endowment Fund and Didia Scholarship of CUG Wuhan. JWH was supported by a Natural Environmental Research Council, UK, IAPETUS DTP research studentship (#NE/L002590/1) hosted at Durham University, UK.

Author contributions
NJS, DS, and JWH designed the study based on samples and detailed geological background provided by JWH. NJS carried out all petrographic investigations followed by sample preparation, quality control of the mineral separates and, Re-Os isotope geochronology and mass spectrometry analyses. DMS and JK provided extensive knowledge of the metallogeny and geodynamic evolution of the Greenland fragment of the North Atlantic Craton. AB carried out sulphur isotope analyses of the aliquots provided by NJS. NJS wrote the manuscript and all other authors contributed comments and edits to the manuscript.

Data and material availability
All data are available in the present publication. Correspondence and material requests should be addressed to corresponding author N.J. Saintilan.

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