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A polymer-based textile thermoelectric generator for wearable energy harvesting

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HIGHLIGHTS

• E-textiles offer wearable sensing and energy harvesting functionality.
• Thermoelectric energy harvesters can convert body heat to electricity.
• Our thermoelectric textile delivers a record 1.2 μW at ΔT = 65 K.
• Adapted thermoelectric models accurately predict the textile device performance.

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ABSTRACT

Conducting polymers offer new opportunities to design soft, conformable and light-weight thermoelectric textile generators that can be unobtrusively integrated into garments or upholstery. Using the widely available conducting polymer:polyelectrolyte complex poly(3,4-ethylenedioxythiophene):poly(styrene sulfonate) (PEDOT: PSS) as the p-type material, we have prepared an electrically conducting sewing thread, which we then embroidered into thick wool fabrics to form out-of-plane thermoelectric textile generators. The influence of device design is discussed in detail, and we show that the performance of e-textile devices can be accurately predicted and optimized using modeling developed for conventional thermoelectric systems, provided that the electrical and thermal contact resistances are included in the model. Finally, we demonstrate a thermoelectric textile device that can generate a, for polymer-based devices, unprecedented power of 1.2 μW at a temperature gradient ΔT of 65 K, and over 0.2 μW at a more modest ΔT of 30 K.

1. Introduction

Electronic textiles (e-textiles) encompass a new class of devices with great potential to transform miniature electronics into truly unobtrusive and ubiquitous systems. Wearable electronics are already part of our daily lives in many forms, ranging from mobile phones and smart watches to pet collars with lights and GPS functionality. Moreover, the use of wearable and even implantable wireless sensors can be expected

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to increase rapidly thanks to low power requirements and emerging applications e.g. remote physical condition monitoring and preventive healthcare [1–3]. For uninterrupted and autonomous operation, such devices are ideally powered by energy scavenged from the wearer or from the environment, rather than relying on batteries which will inevitably require replacement. Energy scavenging devices may convert sunlight, biomechanical movement, friction or body heat to electricity relying on photovoltaic, piezoelectric, triboelectric or thermoelectric principles [4]. For wearable devices to be truly unobtrusive, we propose that electronics in the form of textiles are particularly attractive. Several types of e-textile devices for energy harvesting (Fig. 1) have already been reported. In such devices, multi-layer fibers and fabrics function as photovoltaic, piezoelectric or triboelectric energy converters as demonstrated by several groups Refs. [5–7].

Thermoelectric energy harvesters have the advantage of relying on the wearer’s natural body heat paired with colder surroundings, without any additional requirements such as motion or available sunlight. For example, commercially available Peltier elements can function as wearable thermoelectric energy scavengers, when placed e.g. on a person’s forearm or forehead by means of tape or straps [8–11]. An important conclusion from such studies is that wearable thermoelectric energy harvesting systems will include non-negligible parasitic thermal resistances $K_{par}$. On the hot side, a thermal contact resistance exists between the rough surface of the skin and the smooth and stiff thermoelectric element. On the cold side, heat rejection relies largely on convection. In standard operation a heat sink, i.e. a block of machined aluminum fins, is attached to the generator via thermal paste to increase the surface area available for convection. Clearly, in wearable applications there will exist a trade-off between the size (which is proportional to efficiency) of a heat sink, and the level of discomfort that it causes the wearer.

The large thermal parasitic resistances result in a system where the device performance is largely dictated by the thermal resistance $K_{tc}$ of the thermocouple, and ideally $K_{tc} \gg K_{par}$. This calls for thermoelectric materials with low thermal conductivity $\lambda$. For polymers, $\lambda$ typically ranges from 0.1 to 0.5 Wm$^{-1}$ K$^{-1}$ [12] (c.f. $\lambda_{air} = 0.025$ Wm$^{-1}$K$^{-1}$), compared to $\lambda \approx 1.5$ Wm$^{-1}$K$^{-1}$ [11] for common inorganic thermoelectric materials. Another method to increase the thermal resistance is to increase the device thickness. Consequently, the geometry will play a crucial role for maximized power output. Several studies [9,11,13], which combined comprehensive modeling with experimental work, concluded that for conventional thermoelectric elements used in wearable applications the power output would be optimized by a leg length $L$ much higher than the typical $L \approx 1–2$ mm. Lossec et al. [13] showed that $\Delta P_{\text{optimal}} = 55$ mm for a device with all other properties equal to those of a commercial Peltier element based on bismuth telluride alloys. Such a device would be both, too heavy and too brittle for practical use. On the other hand, polymer materials can be readily shaped into light-weight objects of virtually any size [14].

Currently, there exists a strong interest in the materials science community to develop flexible electronics based on conducting polymers, Polymer materials in general display many attractive properties including ease of processing, flexibility, light weight and – crucial for thermoelectric applications – an inherently low thermal conductivity. Moreover, polymers are usually non-toxic, and even biocompatible – a notable example of this is the electrically conducting polymer:polyelectrolyte complex poly(3,4-ethylendioxythiophene):poly(styrene sulfonate) (PEDOT:PSS) [15]. For thermoelectric applications, a flexible device has the advantage of being conformable to the shape of the human body which can increase the contact area and thus reduce $K_{par}$. Several doped conjugated polymer systems have been explored as thermoelectric materials [16–20], and they can be converted to textiles by various methods e.g. by coating onto a commercial fabric or by spinning into blend or mono-component fibers. The manufacture and properties of electrically conducting organic textile fibers was described in detail in our recent review paper [21].

Thermoelectric polymer-based textiles can also be prepared by coating a conjugated polymer onto fabric substrates. For example, Du et al. prepared an in-plane device with five thermocouples each consisting of a 35 mm × 5 mm cotton fabric strip coated with PEDOT:PSS and a silver wire [22]. They reported a maximum output power $P_{\text{max}} = 12.16$ nW at a temperature gradient $\Delta T = 72$ K, corresponding to 0.03 nW per degree and per leg-pair. Allison et al. used vapor-phase polymerization to deposit p-doped PEDOT (PEDOT:Cl) onto cotton fabrics, and could design an out-of-plane thermoelectric device with two thermocouples, using carbon fiber as the connecting leg, which generated $P_{\text{max}} = 4.5$ nW at $\Delta T = 30$ K (0.075 nW per degree and leg-pair) [23]. With one exception, all reports on organic thermoelectric textile devices use PEDOT [22–34] most commonly combined with PSS as the counterion. Pope and Lekakou used poly(3-hexylthiophene) (P3HT) to coat a cotton yarn [30] which was combined with [6,6]-phenyl-C$_{61}$-butyric acid methyl ester (PCBM) coated n-type yarns and embroidered to form an out-of-plane device with five leg-pairs. The fabricated device produced only $P_{\text{max}} = 0.25$ nW at $\Delta T = 40$ K, presumably due to the low electrical conductivity of the yarns.

With this report we would like to draw attention to the potential of e-textiles as out-of-plane thermoelectric devices. We demonstrate the manufacture and characterization of two 3D textile thermoelectric generators, using materials and processes compatible with existing textile manufacture technologies. Our devices have eight thermocouples and generate $P_{\text{max}} = 1.2$ μW, at a temperature gradient of 65 K (2.3 nW
2. Experimental

2.1. Materials

Approximately 100 m of silk sewing thread (Aurora Silk) was passed through a dye bath prepared with an aqueous PEDOT:PSS dispersion (PH1000 from Heraeus, solid content ~1.3 w%) and 5 vol% ethylene glycol (EG, from Sigma Aldrich). After collection and drying, the coated thread was placed in a vial containing dimethyl sulfoxide (DMSO, from Sigma Aldrich) for 1 h and 20 min, after which the thread was dried again using a heat gun which heated the air to ~100 °C. This roll-to-roll method for continuous coating and dyeing of conducting threads was developed and previously reported by our group [35]. The second conducting thread used in the thermoelectric devices was a silver-plated polyamide embroidery thread (HC12 from Madeira Garnfabrik).

2.2. Manufacture of thermoelectric textiles

The conducting threads were threaded onto a sewing needle and stitched by hand through 9 layers of felted wool fabric (Wadmal, ~1 mm thick, 3.2 g/dm² from Harry Hedgren AB) to form the thermocouple legs. After stitching, a silver-containing paste for textile coatings (PEB74 from Dupont) was applied by rubber stamp printing on the surface of the textile to form electrical connections between the respective legs of each thermocouple. After application the paste was cured by placing the embroidered and printed textile in an oven (CARBOLITE PF30/300C) set to 100 °C, for 10 min. A 30 AWG nickel plated copper hookup wire (2930 from Alpha Wire) with heat resistant fluoropolymer insulation was connected to each node of the thermoelectric device, with the aid of silver paste, to serve as connectors to the instruments.

2.3. SEM characterization

Scanning electron microscopy (SEM) was carried out using a JEOL 7800F Prime, at 2 kV. A thin layer of palladium was sputtered onto the samples prior to microscopy.

2.4. Electrical conductivity

For characterization of electrical conductivity, two samples of each thread were placed on a glass slide, and silver paint (Agar Scientific) was applied to form 10 segments each of 8 mm length. The resistance of each length of thread was measured using a Keithley 2400 source meter unit (SMU), and the conductivity was calculated taking into account the thread diameter as observed with an optical microscope (Carl Zeiss A1).

2.5. Seebeck coefficient characterization

The Seebeck coefficient was measured on at least 4 specimens of each thread, using a digital Seebeck controller SB1000 with a temperature controller K2000 (MMR Technologies) and its low impedance board (gain = 1000). Both ends of the ~5 mm long specimen were attached to the holder with silver paint, and a constantan wire was used as a reference. The measurement was performed at 300 K with a thermal load of 1–2 K.

2.6. Specific heat and thermal conductivity

The specific heat capacity \( c_p \) was characterized in a DSC2 (Mettler Toledo). Following a blank curve correction run, ~5 mg of each sample and ~10 mg of a sapphire reference material was placed in 40 µl aluminum crucibles, and heated to 80 °C at 20 Kmin\(^{-1}\) while recording the heat flow \( \Phi \). Using the materials’ masses \( m \) and the well-defined specific heat capacity of sapphire \( c_{p,sa} \), we could obtain:

\[
\alpha = \frac{c_{p,sa}(\Phi - m\Phi_{sa})}{(m - \Phi_{sa})}
\]

\( \Phi_{sa} \) measured at 23 °C for each material was used as input for the characterization of thermal conductivity, which was carried out with a Hot Disk 2500 S under ambient conditions. This instrument uses a transiently heated plate sensor to simultaneously heat and measure temperature. We characterized our wool fabric, assuming isotropic properties, using a S465 Kapton sensor sandwiched between 7 + 7 layers of wool fabric (area = 5.3 × 5.3 cm\(^2\)) to ensure sufficient volume to accommodate the thermal probing depth. The measurement time was 40 s and heating power 10 mW, resulting in a probing depth of 6 mm. The two electrically conducting threads were expected to display anisotropic characteristics. We prepared rod-like samples from densely packed threads contained in a thermally insulating polypropylene (λ = 0.1–0.2 W m\(^{-1}\)K\(^{-1}\)) tube with an inner diameter of 4.7 mm and lengths of 14.5 mm (PEDOT:PSS coated thread) and 22.6 mm (silver plated thread) respectively. The filled tube was placed in a fitted piece of expanded polystyrene (EPS) to prevent lateral heat conduction. A 7577 Kapton sensor (2 mm diameter) was sandwiched between the thread-packed-cylinder and a second piece of EPS. For the PEDOT:PSS dyed silk thread, the measurement time was 40 s and the heating power was 10 mW, resulting in a probing depth of 14.0 mm. For the silver-plated thread, the measurement time was 10 s and the heating power was 25 mW, resulting in a probing depth of 22.5 mm.

2.7. Thermopile characterization

For characterization of the thermoelectric properties of our devices, we placed the thermoelectric textile on top of a variable temperature hot plate (HP60, Torrey Pines Scientific Inc). Surface mounted K-type thermocouples (Omega Engineering) were placed on the top and at the bottom of the textile, to monitor the surface temperatures via a National Instruments cDAQ 9174 with internal temperature reference. A repurposed CPU-cooler (Hydro Series™ H45) was placed on top of the thermoelectric textiles to maintain a constant cold temperature of 23 °C. Thin sheets of Kapton (50 µm thickness) were placed on both sides of the textile to prevent electric short circuits. The generated voltage was recorded by a Keithley 2400 SMU, which also acted as a variable load by drawing current from the textiles.

3. Results and discussion

3.1. Thermoelectric device structures and materials

When a conducting or semiconducting material, whether it is metal-, metal oxide-, carbon- or polymer-based, is exposed to a temperature gradient \( \Delta T \) this will cause a thermal diffusion and subsequent accumulation of charge carriers. This so-called Seebeck effect results in an electric potential difference \( \Delta V \) in parallel with the \( \Delta T \), and the material’s Seebeck coefficient \( \alpha \) is defined as:

\[
\alpha = \frac{\Delta V}{\Delta T}
\]

The direction of the resulting current will depend on the material’s majority charge carriers, i.e. electrons or holes, and for a p-type material \( \alpha > 0 \), while for an n-type material \( \alpha < 0 \). The smallest unit of a thermoelectric power generator is the thermocouple (tc) (Fig. 2a) constituted of two “legs” – ideally one p-type and one n-type – coupled electrically in series and thermally in parallel so that they both are exposed to the same temperature gradient \( \Delta T_{tc} \) (Fig. 2b–c). The temperature gradient will result in an open circuit voltage potential \( V_{tc} \) over the thermocouple, determined by:

\[
V_{tc} = (\alpha_p - \alpha_n) \Delta T_{tc} = \alpha_c \times \Delta T_{tc}
\]
where $\alpha_p$, $\alpha_n$, and $\alpha_{tc}$ are the Seebeck coefficients of the p-type material, the n-type material and of the thermocouple, respectively. Typically, $\alpha$ is on the order of $\mu$VK$^{-1}$ so in order to produce a relevant voltage, thermoelectric devices will consist of a number of thermocouples connected in series, forming a thermopile (Fig. 2d). Its maximum produced power $P_{\text{max}}$ for a given geometry and under load matching conditions [13], where the internal electrical resistance ($R_i$) is equal to the load resistance ($R_{\text{load}}$), can be calculated as:

$$P_{\text{max}} = \frac{V_{oc}^2}{4R_i} = \frac{(V_n \times m)^2}{4R_i} = \frac{(\alpha_p \times \Delta T \times m)^2}{4R_i} \tag{3}$$

where $m$ is the number of thermocouples in the thermopile and $V_{oc}$ is the open circuit voltage of the thermopile (Fig. 2d–e).

To study the feasibility, and predictability, of such a device, we have aimed to use readily available materials and scalable processing methods for its manufacture (Fig. 3a–b). Previously, our group has developed a roll-to-roll method to coat threads with an ink based on the commercially available conducting polymer:polyelectrolyte complex PEDOT:PSS, resulting in a wash-and-wear resistant electrically conducting thread suitable for sewing or weaving [32,35] (Fig. 3a left). PEDOT:PSS displays a Seebeck coefficient in the range of 10–20 $\mu$VK$^{-1}$ [22,24,28,30,32,36] and for our thread we measure $\alpha = 14.3$ $\mu$VK$^{-1}$ making it useful as the p-type component of a textile device. At present, no air-stable doped n-type polymers are available. Instead, reported organic n-type textiles have been produced using nanocarbon allotropes i.e. doped carbon nanotubes [29,31,37,38], graphene [39] or PCBM [30]. At present the health and environmental risks related to carbon nanomaterials are under investigation, and notably carbon nanotubes were recently added to the SIN- (Substitute It Now) list developed by ChemSec [40,41]. In the absence of organic n-type materials which are with certainty benign, we opted to use a conducting silver-plated embroidery yarn ($\alpha = 0.3$ $\mu$VK$^{-1}$) to connect the p-type legs in our device. To minimize the electrical contact resistance between p-type and silver-legs, a conducting paste for textile print was used to form electrical connections between thermocouples. By hand-embroidering the threads through several layers of wadmal (a felted wool fabric) we were able to design thick out-of-plane thermoelectric textiles (Fig. 3b), in analogy to a conventional thermopile (Fig. 2d). Thus, the thermocouples were incorporated into an insulating wool fabric and when this is worn on a cold day, the body can constitute a hot reservoir and the colder ambient conditions on the outer side of the garment will provide the cold reservoir. The heat flow from the hot to the cold reservoirs is converted to an electric current, and so this textile device provides warm clothing designed for use in a cold climate, with the added functionality of unobtrusive energy scavenging (Fig. 3c).

### 3.2. The impact of thermal contact resistance

Under practical conditions there exist thermal contact resistances between a thermocouple and the cold and hot reservoirs, respectively. Assuming conductive heat transfer, the effective $\Delta T_{tc}$ driving the

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**Fig. 2.** (a) Schematic representation of a thermocouple (tc) placed between a hot and a cold reservoir. The temperature gradient induces charge carrier diffusion resulting in an electric voltage $V_{oc}$ and a drift current $I$. The light-blue and light-yellow fields represent thermal contact resistances. (b) The thermal contact resistances introduce a temperature loss resulting in a temperature gradient over the length $L$ of the thermocouple ($\Delta T_{tc}$) which is smaller than the temperature difference ($\Delta T$) between the hot and cold reservoirs. (c) Equivalent circuit representing the thermal resistances in a thermocouple, where $K_p$ and $K_n$ represent the contact resistances on the cold and hot side and $K_{ntc}$ and $K_{pntc}$ are the thermal resistances of the p-type, n-type and insulating materials. (d) Schematic representation of a wearable thermopile consisting of several thermocouples connected electrically in series and thermally in parallel. (e) Equivalent electrical circuit of the thermopile where $V_{oc}$ is the open circuit voltage, $R_i$ and $R_{load}$ are the internal resistance and the load resistance, $I$ is the current and $V_{out}$ is the voltage measured at the device terminals. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)
thermoelectric generator can be predicted by:
\[
\Delta T_s = \frac{K_a}{K_c + K_{p/c} + K_{n/c}} \Delta T
\]  

where \( \Delta T \) is the available difference in temperature between the hot and cold reservoirs, \( K_c \) is the thermal resistance of the thermocouple, and \( K_{p/c} \) and \( K_{n/c} \) represent the thermal contact resistances between the generator and the ambient conditions (cold reservoir) and the generator and the body (hot reservoir). In turn, \( K_c \) is the combination of the thermal resistances of the p-type material \( (K_p) \), the n-type material \( (K_n) \) and the thermal insulating material filling the void between the thermocouples \( (K_{ins}) \). These thermal resistances can be regarded as coupled in parallel (Fig. 2c), and the resulting \( K_c \) is defined by:
\[
\frac{1}{K_c} = \frac{1}{K_p} + \frac{1}{K_n} + \frac{1}{K_{ins}}
\]  

The thermal resistance of a given component is determined by its thermal conductivity \( \lambda \), its length \( L \) and its cross-section area \( A \), as in:
\[
K = \frac{L}{\delta \times A}
\]  

Likewise, the electrical resistance is defined by a material’s electrical conductivity \( \sigma \) and geometry i.e.:
\[
R = \frac{L}{\sigma \times A}
\]  

In addition, electrical contact resistances will be present at all interfaces of the thermocouple. Clearly, the power that a thermoelectric generator can provide is determined not only by material properties but also by geometric considerations. Previously reported wearable thermoelectric generators inevitably have a heatsink attached, and for efficient cooling the devices are positioned so as not to be buried under clothing e.g. on the forehead or on the naked arm [3, 9–11, 13]. This approach limits the use of wearable devices to warm conditions, resulting in a low available \( \Delta T \). Moreover, for a wearable energy harvesting device to be attractive to wear, i.e. to be unobtrusive and comfortable, we propose that the use of conventional heatsinks is simply not an option. Instead, we choose to integrate our device into a textile of low thermal conductivity, and to optimize its geometry (in particular, its thickness) in order to minimize the impact of thermal contact resistances. According to equations (4)–(6), a high \( \Delta T_s/\Delta T \) ratio is promoted by:

i. Low thermal contact resistances
ii. Low thermal conductivity of the materials used
iii. A low fill factor (assuming that \( K_{ins} < K_{p/c} \))
iv. A large leg length (= device thickness).

Factor (i) is conventionally met by the application of thermal pastes and, on the hot side of the device, aluminum fin heat sinks. Since this is not a viable option for textile wearable devices, we focus instead on the remaining factors. The thermal conductivity \( \lambda \) is defined by thermal diffusivity \( \delta \), material density \( \rho \) and specific heat \( c_p \) as in:
\[
\lambda = \delta \rho c_p
\]  

Because a textile is a porous composite consisting largely of air and some humidity, a textile fabric can display an order of magnitude lower \( \lambda \) compared to its constituent fibers [42]. We have carefully characterized the thermal conductivities of our materials. In our PEDOT:PSS coated silk threads, each thread is composed of many silk filaments placed in parallel along the thread axis, surrounded by the conducting coating (Fig. 4a, left). Our second conducting thread consists of many individual silver-plated polyamide fibers, held together in a 2-ply twist (Fig. 4a, center). Since the textile fibers are aligned (also on a macro-molecular scale) we assume that they display anisotropic thermal properties [12]. The thermal gradient in our out-of-plane device will be

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**Fig. 3.** (a) Photographs of the materials used in our thermoelectric textile: (left) a PEDOT:PSS coated silk thread, (center) a silver plated polyamide thread and (right) several layers of felted wool fabric. Scale bars are 10 mm. (b) Schematic of the embroidery process, showing the position of the threads in a cross-section of the wool fabric. Each cluster of threads represents individual p-type (blue) and n-type (grey) legs. The legs will be electrically connected in series by application of silver paste on the surface of the fabric, to form a thermopile. (c) Schematic illustration of a flexible textile thermopile placed on the skin. The high thickness and low thermal conductivity of the textile ensures its function as an insulator as well as a thermoelectric generator. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)
parallel to the axis of the threads, consequently we prepared the thread samples for thermal characterization by repeatedly threading them through a polypropylene cylinder (Fig. 4b) until the cylinder was densely filled. For characterization, we placed the sensor on top of this cylinder (Fig. 4c, left) with a thick insulator, in turn, on top of the sensor. The felted wool fabric, instead, consists of textured (‘wavy’) wool fibers arranged in a random non-woven architecture (Fig. 4a, right). This fabric can be assumed to display isotropic thermal properties and was characterized as a bulk material, with the sensor instead sandwiched between two thick multi-layer wool samples (Fig. 4c, right). We found that for our PEDOT:PSS coated silk thread, $\lambda = 0.18 \text{ Wm}^{-1}\text{K}^{-1}$, the silver plated polyamide thread has a higher thermal conductivity of $\lambda = 0.47 \text{ Wm}^{-1}\text{K}^{-1}$ and the porous wool fabric displays a low $\lambda = 0.056 \text{ Wm}^{-1}\text{K}^{-1}$ (see Table 1 for details). Consequently, factor (ii): low thermal conductivity is fulfilled by our materials.

We chose a design for our embroidered device, which mimics that of conventional thermopiles. In order to generate a high power, the thermopile should ideally have a low internal resistance. Consequently, a high thread count in each thermocouple leg is advantageous, which means that the area of each leg needs to be relatively large. In our device (Fig. 5) we design each leg to be 8 mm by 8 mm, with a spacing of 3–3.5 mm between them to ensure that no electric short circuiting will occur. Our thermopile will have 8 thermocouples connected in series. After initial embroidery trials we found that we can incorporate 133 threads per leg. The total area of each thermocouple ($tc$) will be $11 \text{ mm} \times 23 \text{ mm} = 253 \text{ mm}^2$. Note that each leg will consist of both conducting thread and insulating wool (c.f. Fig. 3b), so in order to calculate the leg cross-section areas we use the thread count $N$ multiplied by the cross-section areas of the n-type and p-type threads, $A_n$ and $A_p$, respectively. The total area of insulating material per thermocouple $A_{ins} = A_{tc} - N (A_p + A_n) = 235 \text{ mm}^2$, and the fill factor $FF = N (A_p + A_n)/A_{tc} = 0.07$. In conclusion, a very low fill factor (compared to ~25% for commercial thermopiles [11]), c.f. Point (iii): a low fill factor above, is fully achievable with textiles, as the active components can be embedded in cloth and do not need to be self-supporting.

Moreover, in accordance with point (iv): a large leg length, the textile format is ideal for designing thick devices whereas an inorganic device of $> 5 \text{ mm}$ thickness would be perceived as heavy and bulky. Textiles are commonly used in centimeter thick configurations, for example in upholstery and mattresses, as well as in outdoor jackets, winter boots and sleeping bags for use in cold climates. The thermal contact resistance $K_b$ between the human body and the thermoelectric module is determined by several factors including the thermal conductivity of the skin, the hardness and the topology of the skin, location on the body and the pressure exerted on the skin by the device. The heat transfer coefficient $h_b$ between the human body and a thermoelectric module has been estimated to be 20 - 100 Wm$^{-2}\text{K}^{-1}$ [11,13,43]. The cold side thermal contact resistance is determined by natural convection

![Fig. 4.](image-url)

(a) Scanning electron micrographs of (left) a cross-section of the PEDOT:PSS coated silk thread, displaying the individual silk filaments in the thread, (center) a top view of the silver plated thread and (right) a top view of the wool fabric. Scale bars are 100 μm. (b) Photographs of the sample preparation process for characterization of anisotropic thermal conductivity, where (left) the thread was repeatedly stitched through a polypropylene cylinder to form a (right) rod shaped sample of densely packed threads. Scale bars are 10 mm. (c) Schematic illustrations of the setup for characterization of thermal conductivity of (left) anisotropic samples and (right) isotropic samples.
and radiation to the colder environment. The heat transfer coefficient to ambient \(h_{\text{amb}}\), without use of a heat sink, is estimated to be relatively low at 6 - 11 Wm\(^{-2}\)K\(^{-1}\) [13,43]. The heat transfer coefficient relates to the thermal resistance \(K\) and the contact area \(A\) as in eq. (9):

\[
K = \frac{1}{h_{\text{amb}} A}
\]

(9)

Based on the “best case” and “worst case” scenarios from literature we can assume that the contact resistances, per thermocouple, of our textile energy harvester will vary between \((K_a + K_b)_{\text{min}} = 390\ \text{KW}^{-1}\) and \((K_a + K_b)_{\text{max}} = 860\ \text{KW}^{-1}\). From the material properties, we use eqs. (5) and (6) to calculate the thermocouple’s intrinsic thermal resistance to \(K_{\text{th}} = L \times 5 \times 10^4\ \text{KW}^{-1}\).

3.3. The impact of electrical contact resistance

In addition to the intrinsic electrical resistance of our materials, there will exist electrical contact resistances between each leg and thermocouple. We estimate the contact resistance to be about 0.5 \(\Omega\) per thermocouple. We then proceeded to characterize the bulk electrical conductivity (Table 1) of our threads. It is quite common in e-textile literature to report the surface resistance in terms of \(\Omega \cdot \text{cm}^{-1}\) or as Scm\(^{-1}\) by using an estimated thickness of the conducting layer. We find however that it is more relevant, for the present application, to use the bulk volume conductivity \(\sigma_b\), therefore we calculate the cross-section area of the conductor using the average thread diameter as observed by optical microscopy. Consequently, a large part of the characterized volume will be constituted by insulating polymer, and \(\sigma_b\) will be relatively low compared to the \(\sigma\) of the conducting component itself (up to 2\times10^3 Scm\(^{-1}\) for post-treated PEDOT:PSS [36]). To ensure that we access the full circumference of the thread, we apply conducting silver paint at several points along the length of thread samples and measure the electrical resistance between these points, in a two-point configuration. We measure \(\sigma_b = 43\ \text{Scm}^{-1}\) for our PEDOT:PSS coated threads, and \(\sigma_b = 1600\ \text{Scm}^{-1}\) for the silver plated polyamide threads.

![A schematic outline of a textile thermopile, top view. Silver paste is applied on top of the embroidered textile to form the series electrical connection between thermocouples.](image)

By combining this information with equations (2)–(9), we can predict the generated power as a function of the leg length \(L\), at a given \(\Delta T\). We find that the contact resistances, both thermal and electrical, have a significant influence on the predicted thermoelectric voltage and power (Fig. 6). Note that for the purely hypothetical case of zero thermal contact resistance, assuming steady state conditions, \(L\) should be as small as possible as this results in minimal internal electrical resistance. However, because the potential difference \(V_{\text{oc}}\) is proportional to the temperature gradient at the textile boundaries, the addition of thermal contact resistances results in a significantly suppressed electric potential at low \(L\) (c.f. eq. (4)). In fact, only at \(L > 100\ \text{mm}\) does the open circuit voltage approach the value predicted by (eqs. (2) and (3)) \(V_{\text{oc}} = (\sigma_p - \sigma_b) \times m \times \Delta T = 14 \times 10^{-4} \times 8 \times 65 = 7.28\ \text{mV}\) (Fig. 6a–b, grey lines with triangle symbols) for our device. But as \(L\) increases, so does the internal electrical resistance, resulting in a reduction of generated power (Fig. 6a–b, solid lines). For the most optimistic case in terms of contact resistances (where \(h_a + h_b = 111\ \text{Wm}^{-1}\text{K}^{-1}\), \((K_a + K_b)_{\text{min}} = 390\ \text{KW}^{-1}\) and the electrical contact resistance is 0.25 \(\Omega\) per thermopile, the maximum power output \(P_{\text{max}}\) is found to be 0.8 \(\mu\text{W}\) at \(L = 15\ \text{mm}\) (Fig. 6b). In our worst case scenario instead where \(h_a + h_b = 26\ \text{Wm}^{-1}\text{K}^{-1}\) the thermal contact resistances are \((K_a + K_b)_{\text{max}} = 860\ \text{KW}^{-1}\) and the electric contact resistance is 1 \(\Omega\) per thermocouple, \(P_{\text{max}} = 0.3\ \mu\text{W}\) at the optimal \(L = 40\ \text{mm}\) (Fig. 6a). In our lab setup, we may expect the thermal contact resistances to be relatively low and we therefore design our device according to the more optimistic predictions. As embroidery will become increasing difficult with an increasing number of fabric layers, we limit our textile device thickness to 10 mm.

3.4. Characterization of our thermoelectric textile generator

We proceeded to manufacture an out-of-plane thermopile by hand-embroidering our threads through 9 layers of wool fabric. After embroidery, a silver-based conductive coating developed for textiles was used to connect the legs to form an electrical series circuit (Fig. 7a), and two thin copper wires were attached to serve as electrodes. We measured the internal resistance \(R_i\) of our device to be 13 \(\Omega\). The theoretical value, calculated from geometry and material properties (eq. (7)) would be \(R_i = 3.4\ \Omega\). We conclude that the electrical contact resistance is 1.2 \(\Omega\) per thermocouple (\(\approx 9.6\ \Omega\) for the thermopile), which means that the contact resistance is about 3 times higher than the intrinsic resistance of our textile. We proceeded to characterize the thermoelectric performance of our textile by placing it on top of a hot plate, with a repurposed CPU-

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**Table 1**

| Material                        | diameter (µm) | \(\rho\) (kgm\(^{-3}\)) | \(\sigma_b\) (Scm\(^{-1}\)) | \(\alpha\) (µVK\(^{-1}\)) | \(c_p\) (Jg\(^{-1}\)K\(^{-1}\)) | \(\lambda\) (mm\(^{-2}\)s\(^{-1}\)) | \(\lambda\) (Wm\(^{-1}\)K\(^{-1}\)) |
|--------------------------------|--------------|------------------------|--------------------------|------------------------|-------------------------------|---------------------------------|----------------------------------|
| Felted wool                    | n.a.         | 290                    | n.a.                     | 1.14 ± 0.07            | 1.46                          | 0.14                            | 0.056                            |
| PEDOT:PSS coated threads       | 230          | 840                    | 43 ± 10                  | 14.3 ± 0.07            | 1.46                          | 0.14                            | 0.18                             |
| Silver-plated threads          | 350          | 930                    | 1600 ± 300              | 0.3 ± 0.1              | 1.22                          | 0.42                            | 0.47                             |

n.a. = not applicable.
cooler which holds a constant temperature of 23 °C on top of the textile device (Fig. 7b). We did not apply any thermal paste or other means to enhance thermal contact. We increased the temperature of the hot plate to stepwise increase ΔT and recorded the generated power while drawing an increasing current, i.e. the source-measure unit acted as a variable load to our textile device. The open circuit voltage $V_{oc}$ was taken as the extrapolated voltage at $I \to 0$. We find that $V_{oc}$ linearly increases with ΔT and is only slightly lower than the $V_{oc}$ predicted by eqs. (2) and (3) (Fig. 7c). From this we could estimate that the ratio of thermal resistances in our setup is $K_{tc}/(K_a + K_b + K_{tc}) \approx 0.9$ which is equal to a $h_a + h_b = 246 \text{ Wm}^{-2}\text{K}^{-1}$. This means that the heat transfer coefficient in our lab setup is about twice as high as that of the "best" wearable system. With this information about the electrical and thermal contact resistances in our system, we could predict the generated power as a function of temperature gradient (Fig. 7d, solid lines), and found that the calculations are a good match with the experimental data.
Clearly, the device can be further optimized in several ways. One option is to use the materials more efficiently by taking into account their different electrical conductivities, as previously described by We et al. [44] When the sum $A_n + A_p = \text{constant}$ then $P_{\text{max}}$ occurs at:

$$A_n = \sqrt{\frac{\lambda_n}{\lambda_n + \lambda_p}}$$  \hspace{1cm} (10)

In our textile device, $A_n$ and $A_p$ are directly proportional to the thread count $N$ of each leg and their respective diameters $D$, so that:

$$N_p = \sqrt{\frac{\sigma_p \times \lambda_p}{\sigma_p \times \lambda_n} \times \left(\frac{D_p}{D_n}\right)^2} = \sqrt{\frac{43 \times 0.18}{1557 \times 0.47} \times \left(\frac{320}{350}\right)^2} \approx 0.04$$  \hspace{1cm} (11)

Based on this result, we designed a second thermoelectric textile with 8 thermocouples and a thickness of 10 mm, modifying only the thread counts to be $N_p = 398$ and $N_n = 18$ (in the first device $N_p = N_n = 133$), resulting in the device design in Fig. 8a. We manufactured this second device, and measured an internal resistance $R_i$ of 10 $\Omega$ whereas according to calculations $R_i = 1.4$ $\Omega$. Again, the internal electrical resistance of our textile device is mainly constituted by contact resistance. We characterized the second device as described previously. As expected, the thermoelectric potential difference again closely follows the predicted $V_{oc}$ in accordance with eqs. (2) and (3) (Fig. 8b), confirming that the thermal contact resistances are of the same magnitude as during the characterization of our first device. The lower internal resistance of our second textile generator resulted in an increase in generated power (Fig. 8c, purple circles), with 1.22 $\mu$W generated at $\Delta T = 65$ °C, again a close match to the predicted trend (Fig. 8c, solid blue line).

### 3.6. Future outlook

The optimization of thermoelectric generators involves a complex relationship between material properties and geometric considerations, and we find it reassuring that traditional calculations, once they have been modified to include thermal and electrical contact resistances, can accurately predict the performance of e-textile devices. In this work, we have not used thermal paste but instead a high device thickness to reduce the negative impact of thermal contact resistances on device performance. Suarez et al. [11] pointed out that thin flexible heat sinks (fin height $= 1-3$ mm) made from polymers with a thermal conductivity of more than 5 W m$^{-1}$K$^{-1}$ could be a viable option for wearable applications, especially when used under windy conditions. Such polymer materials can be manufactured both, from highly aligned polymers as well as by adding fillers to a polymer matrix [12]. We anticipate that future developments of textile thermoelectrics will also include fabrics designed to mimic conventional heat sinks, i.e. with a large surface area.

In addition, the electrical contact resistance has a considerable influence on the power generated. Due to the basic principle of a thermoelectric device, where many legs of different materials are connected in series, electrical contact resistance will always be present in the circuit. Techniques to minimize them warrant further study, e.g. Kirihara et al. [45] reported that the contact resistance at a PEDOT:PSS-metal junction could be significantly reduced by post treatment with EG or DMSO. For the interested reader, we also refer to a recent review paper on interconnects in e-textiles from Agcayazi et al. [46].

### 4. Conclusions

- E-textile materials can be used to design out-of-plane thermoelectric generators with an unusually large thickness, low fill factor and low thermal conductivity – all of which will increase the thermal gradient over the device.

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**Fig. 8.** (a) Schematic outline of our second textile thermopile, with optimized leg areas. (b) Measured open circuit voltage $V_{oc}$ (circles) as a function of the temperature gradient $\Delta T$ between the hot plate and the cooler, and calculated data for $V_{oc}$ as a function of $\Delta T$ where $\Delta T_{oc} = \Delta T$ (solid line) and $\Delta T_{oc} = 0.9 \cdot \Delta T$ (dashed line). (c) Measured generated power (circles, red for the 1st thermopile, purple for the 2nd thermopile) as a function of $\Delta T$ and calculated generated power (solid lines, grey for the 1st thermopile, blue for the 2nd thermopile) assuming $\Delta T_{oc} = 0.9 \cdot \Delta T$ and electrical contact resistance of $1.2 \Omega$ (1st thermopile) or $1.1 \Omega$ (2nd thermopile) per thermocouple. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)
• Standard thermoelectric models can be used to accurately predict the performance of e-textile devices, provided that the electrical and thermal contact resistances are included in the model.

• We have used such models to design a thermoelectric textile device which could produce a power >1 mW at ΔT = 65 K, exceeding the previously reported performance achieved by polymer-based textile devices.

Author Statement
A.L. planned the project, developed the instrumentation for textile thermoelectrics characterization, developed the models applied for e-textiles, performed the calculations, wrote the manuscript. Y.T. manufactured and characterized the textile thermoelectric devices. S.D. developed the method to manufacture the polymer-based conducting sewing thread, assisted in manufacture and characterization of the materials. C.M. planned the project, co-wrote the manuscript.

Declaration of competing interest
The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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