Nanocomposite plasters for the treatment of superficial tumors by chemo-photothermal combination therapy

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Introduction: Novel nanomedical systems are being developed as multiple therapeutic modalities because the combinational therapies for cancer on a single platform can have larger chance to address tumor heterogeneity and drug resistance than any mono-therapeutic modality.

Methods: In this study, photothermal therapy (PTT) and chemotherapy (CT) were combined to treat squamous cell carcinoma by using a novel type of noninvasive plaster composed of carboxylated-reduced graphene oxide (rGO–COOH), gold nanorods (Au NRs), and doxorubicin (DOX). Firstly, DOX was loaded onto rGO–COOH to form DOX_rGO–COOH. Then, the obtained DOX_rGO–COOH and Au NRs were co-assembled to obtain nanocomposite multilayer. rGO–COOH and Au NRs were combined together to obtain high light-to-heat conversion efficiency. Using them as photothermal agents for PTT and using DOX in rGO–COOH as an anticancer drug for CT, their synergistic combination therapy could be applicable practically.

Results: As a result, DOX_rGO–COOH/Au NRs showed higher photothermal effects than that showed by rGO–COOH or Au NRs alone. It also showed higher therapeutic effects than DOX_rGO–COOH (for CT) or Au nr (for PTT) alone. Moreover, the system can repeatedly produce heat and simultaneously stimulate the release of the encapsulated anticancer drug into the tumor upon being irradiated by near-infrared laser. In vivo experiments demonstrated that the squamous cell carcinoma-bearing mice treated with DOX_rGO–COOH/Au NRs were healthy for more than 60 days without tumor recurrence.

Conclusion: The as-developed DOX_rGO–COOH/Au NRs plaster could be an effective, convenient, and noninvasive treatment option for treating superficial tumors.

Keywords: nanocomposite plaster, carboxylated-reduced graphene oxide, doxorubicin, Au nanorods, superficial tumor

Introduction
Cancer is one of the most common life-threatening diseases and a leading cause of deaths worldwide. Although chemotherapy (CT) remains the major therapeutic approach for treatment of most of the cancers, photothermal therapy (PTT) has efficiently treated superficial tumors in preclinical studies. For example, it is currently being tested in clinical trials for treating superficial head and neck cancer by Nanospectra Biosciences, Inc. PTT, also known as PT ablation or optical hyperthermia, has been actively explored as a promising alternative or supplement to conventional cancer treatment approaches due to its remote controllability, minimal invasion, easy applicability, and low systemic toxicity. It employs PT agents to generate sufficient heat to raise the local temperature under specific light exposure. The heat can cause irreversible...
cellular damage leading to cell death. Moreover, numerous reports have found that mild PT heating can enhance the cellular uptake of chemotherapeutic drugs.5,6 Thus, the combination of PTT and CT has been proposed to yield a synergistic effect with better antitumor efficiency and more clinical applications.7-10 Some composite systems have been prepared including mesoporous silica nanoparticles,11,12 graphene quantum dots,13 metal–organic framework14 for combinational PTT and CT cancer treatment. Li et al reported a carrier-free theranostic system co-assembled with chemotherapeutic drug, epirubicin (EPI), and photothermal agent, indocyanine green (ICG) for dual-modal imaging-guided synergistic chemo-PTT.15

For this combinational treatment, the development of PT agents, which can absorb and convert light to heat efficiently, would be a key point.16 Various PT transducers, including organic compounds,17-19 metal nanostructures,20-23 copper chalcogenide semiconductors,24-26 and carbon-based materials,27-30 have been reported for the therapy of tumors. Among the carbon-based materials, graphene oxide (GO), particularly reduced GO (rGO), has become the most competitive PT agents due to its optical transparency, high chemical stability, and effective light-to-heat conversion.31,32 However, the low quantum efficiency of rGO and its broad absorption spectrum render it less sensitive to specific wavelengths.33,34 To overcome these limitations and to enhance the PT effect, rGO-Au composite systems have been proposed.35,36

Gold (Au) nanostructures including aggregates of colloidal particles, nanoshells, nanocages, nanorods, and nanocrosses have received increasing attention in the field of PTT in recent years due to the fact that their localized surface plasmon resonance (SPR) confers them with the capability to absorb light at a specific wavelength and efficiently convert the photon energy into heat.37 Importantly, the SPR peaks can be easily tuned by altering their size, shape, structure, or a combination of these parameters, for example, unlike spherical nanoparticles, which absorb light in the visible region, Au nanorods (Au NRs) shift the excitation wavelength to the near-infrared (NIR) region. The use of NIR light is preferred for PTT because water, melanin, and hemoglobin have absorption minima between 700 and 900 nm, and the light at this wavelength is most likely to pass directly through the tissues without significant heat generation.38

Despite their potential therapeutic effects, treatments with PTT nanomaterials that are delivered to tumors either intravenously or intratumorally, are often repeated because of incomplete tumor cell ablation and tumor recurrence resulting from localized hyperthermia and rapid clearance of the nanomaterials from the tumor sites. Multiple and frequent injections of these nanomaterials may cause adverse side effects and patient discomfort.

Here, we developed a type of nanocomposite plaster with Layer-by-Layer (LbL) assembly technique aiming to use it as a novel strategy for superficial cancer treatment. It can repeatedly produce heat, and simultaneously, release the encapsulated anticancer drug into the tumor upon NIR laser. The LbL method for fabrication of thin films has become a subject of intensive research for various biomedical applications, from drug delivery,39,40 cellular adhesive surfaces41,42 to tissue regeneration.43,44 Until now, various kinds of materials, such as nanoparticles,45 dendrimers,46 charged small molecules,47 DNA origami,48 carbon nanotubes,49 fullerences,50 and graphene,51 have been incorporated in the LbL films along with polyelectrolytes due to versatility and simplicity of the method. For example, we have constructed functional thin composite films by incorporating rGO or GO into the LbL multilayer for potential biomedical applications, such as biosensing52,53 and biointerface construction.54,55

In this study, doxorubicin (DOX) has been used as a chemotherapeutic agent and adsorbed onto carboxylated-reduced graphene oxide (rGO–COOH) forming DOX_rGO–COOH. Then, the formed DOX_rGO–COOH was co-assembled with Au NRs to fabricate the DOX_rGO–COOH/Au NRs multilayer. We studied the synergistic PTT and CT effects of the multilayer in vitro and in vivo, as shown in Scheme 1. Its toxicity has been evaluated before and after NIR light radiation in human skin squamous cancer cells, A-431. The therapeutic efficacy of the film has been investigated in vivo using squamous cell carcinoma-bearing mice. Particularly, we have revealed that the combination of DOX_rGO–COOH and Au NRs in the composite multilayer considerably enhances the therapeutic effects, thus allowing it to act as a superior therapeutic plaster against skin cancer.

Materials and methods

Chemicals and materials

Polyethyleneimine solution (PEI, 50% in H₂O), DOX hydrochloride, and gold (III) chloride trihydrate (HAuCl₄·3H₂O) were purchased from Sigma-Aldrich (St Louis, MO, USA). Expandable graphite 8099200 was purchased from Qingdao BCSM. Co. Ltd. (Laixi, China). Cetyltrimethylammonium bromide (CTAB), NaBH₄, AgNO₃, HCl, and NaOH were gained from Tianjin Damao Chemical Reagent Co. Ltd. (Tianjin, China). 98% H₂SO₄, 30% H₂O₂, and KMnO₄ were purchased from Shanghai Chenyun Chemical and Engineering Company (Shanghai, China). All the chemical
Reagents were used as received without further purification. A diode laser (808 nm) was bought from Beijing Optoelectronic Technology Co. Ltd. (Beijing, China). The water used in this work has been purified using Millipore Milli-Q system (Burlington, MA, USA).

**Synthesis of Au NR**

Au NR were synthesized by traditional seed-mediated method. Firstly, gold seeds were prepared by adding 250 µL 0.01 M HAuCl₄ into 9.75 mL 0.1 M CTAB, followed by gentle mixing. Subsequently, 600 µL 0.01 M ice-cold NaBH₄ was added immediately. The above system was incubated for 2 hours at 28°C–30°C and left undisturbed until further use. Then, nanorod growth solution was prepared by mixing 2 mL 0.01 M HAuCl₄, 400 µL 0.01 M AgNO₃, 800 µL 0.1 M HCl, and 40 mL 0.1 M CTAB. Finally, 200 µL gold seed solution was added when the growth solution became colorless, and the system was stored at 30°C overnight. The Au NRs were collected and purified by centrifugation (10,000 rpm, 10 minutes), and the pellet was resuspended into Millipore water.

**Preparation of rGO–COOH**

GO was synthesized by modified Hummers’ method using graphitic powders. Firstly, 0.5 g graphitic powder and 25 mL concentrated H₂SO₄ were mixed under ice bath and stirred for 2 hours. Then, 2.5 g KMnO₄ was added, heated to 35°C for 4 hours. Next, with the addition of 50 mL water, the reaction was maintained at 10 minutes at 98°C. Further, 150 mL water was added into the system to dilute it, and 30% H₂O₂ was added to stop the reaction. Finally, the mixture was filtered through 0.45 µm filter membrane and washed repeatedly until the pH of the filtrate was within the range of 5–6. The filter was redispersed into water and stored until further use. Next, GO-COOH was obtained as described previously. Briefly, NaOH (1.4 g) and Cl–CH₂–COOH (1 g) were added to GO (2 mg/mL, 10 mL) suspension and sonicated for 2 hours at 80°C. The reaction can convert –OH on GO to –COOH via conjugation of acetic acid moieties and can partially reduce GO into rGO. The resulting rGO–COOH solution was quenched with HCl (20%), washed until neutral pH was obtained, and purified by repeated centrifugation (4,000 rpm, 20 minutes).
Loading of DOX onto rGO–COOH

DOX was loaded on to rGO–COOH by mixing DOX solution (2 mg/mL) with rGO–COOH dispersion (1 mg/mL). The absorption time was 12 hours under dark conditions. DOX_rGO–COOH was collected by centrifugation at 8,000 rpm for 10 minutes. After being washed twice by water, drug loading was evaluated using ultraviolet–visible spectrophotometry (UV-Vis) absorbance at 488 nm. The loading efficiency was calculated by the following equation:

\[
DOX_{\text{loaded}} \% = \frac{DOX_{\text{total}} - DOX_{\text{free}}}{DOX_{\text{total}}} \times 100\%
\]

Fabrication of DOX_rGO–COOH/Au NRs multilayer

DOX_rGO–COOH/Au NRs multilayer was fabricated using the LbL assembly technique. In this study, some substrates, such as glass slides, quartz wafers, and clinical nonwoven fabric, were used for different characterization.

To increase the hydrophilicity of the nonwoven fabric, it was treated with plasma cleaner for 5 minutes before assembly. Multilayers were constructed by alternate deposition of DOX_rGO–COOH and Au NRs onto the substrate. Each deposition was performed with the help of vacuum drying, and the deposition cycles were determined by the desired number of layers. While in the case of glass slides and quartz wafers, they were firstly treated with piranha solution (98% H\(_2\)SO\(_4\); 30% H\(_2\)O\(_2\) = 7:3) overnight to obtain the hydroxylated surface. Then, PEI (1.5 mg/mL, in 0.15 M NaCl) was adsorbed as the first layer for the successive assembly. Steps were repeated to obtain a desired number of layers. In this study, DOX_rGO–COOH/Au NRs with four bilayers, defined as (DOX_rGO–COOH/Au NRs), was used for in vitro and in vivo evaluation. It was calculated by mass evaluation that 0.39 µg/mm\(^2\) of rGO–COOH and 0.44 µg/mm\(^2\) of Au NRs were deposited in each cycle averagely.

Release of DOX from DOX_rGO–COOH/Au NRs multilayer

DOX release experiments were performed in PBS with and without 808 nm laser radiation (0.7 W cm\(^{-2}\)), respectively. In both the cases, DOX_rGO–COOH and DOX_rGO–COOH/Au NRs were immersed in PBS. The supernatant was examined at different time points by measuring the absorbance at 488 nm. This experiment was performed three times for each sample. The cumulative release of DOX was determined by the following equation:

\[
DOX_{\text{released}} \% = \frac{DOX_{\text{released}}}{DOX_{\text{initial}}} \times 100\%
\]

Characterization

The size and distribution of Au NRs were evaluated by transmission electron microscope (TEM) JEM-1011 (JEOL, Tokyo, Japan). The zeta-potentials of rGO–COOH and Au NRs were determined by Zetasizer Nano ZS ZEN 3600 (Malvern Instruments, Malvern, UK). In this study, all UV-Vis absorption spectra were obtained from UV3600 spectroscope (Shimadzu Corp., Kyoto, Japan), and Fourier transform infrared spectroscopy (FTIR) spectra were obtained from Tensor-27 (Bruker Corporation, Billerica, MA, USA). The surface morphology of DOX_rGO–COOH/Au NRs was observed using Nanoscope IIIa apparatus (Digital Instruments, Santa Barbara, CA, USA). Temperature change of all systems was monitored using DAE-905k thermometer (SENDAE, Beijing, China).

Cell culture

Human skin squamous cancer cells A-431 were purchased from Center for Type Culture Collection of Wuhan University (Wuhan, China). A-431 cells were cultured in Eagle’s minimum essential medium (EMEM) supplemented with 10% (v/v) fetal bovine serum, 1.0 mM pyruvate sodium, and 100 units/mL penicillin/streptomycin at 37°C in a CO\(_2\) incubator (Thermo Fisher Scientific, Waltham, MA, USA) with 5% CO\(_2\). Multiplication of cells in the logarithmic phase was conducted with 0.05% trypsin.

Cell viability

Cell viability was evaluated using a live/dead assay kit (Thermo Fisher Scientific). Seven experimental groups, namely tissue culture polystyrene (TCP) as a control group, blank glass slide, rGO–COOH on glass slide, Au NRs on glass slide, rGO–COOH/Au NRs on glass slide, DOX_rGO–COOH on glass slide, DOX_rGO–COOH/Au NRs on glass slide, were investigated. Briefly, cells (1.5x10\(^4\) cells/mL) were seeded in each well of a 24-well cell culture plate. Each group was divided into two parts, one part was cultured normally (without NIR laser radiation) and the other under 808 nm laser radiation (0.7 W cm\(^{-2}\), 10 minutes). For those systems under radiation, cells were cultured for another 6 hours before adding live/dead assay solution containing
calcein-AM and propidium iodide (PI). The fluorescence images were obtained with confocal laser scanning microscopy (CLSM) FV 1000 (Olympus Corporation, Tokyo, Japan). Quantitative analysis of cell viability was determined by CCK-8 assay for all the groups.

Animal experiments
Animal experiments were performed following the guidelines and animal ethical regulations published by the Ministry of Science and Technology of the People’s Republic of China. The experiments in this study were approved by Animal Care and Use Committee of Institute of Process Engineering, Chinese Academy of Sciences, where the experiments were conducted. Four-week-old female BALB/c nude mice weighting 20–25 g supplied by State Key Laboratory of Biochemical Engineering (Beijing, China) were used as animal models for the evaluation of in vivo therapy effects. A-431 cells (1.0 × 10^7) in EMEM were injected into the right hind leg of mice. When the tumor volume grew to approximately 120 mm^3, the mice were divided randomly into seven groups: the normal growth control group, the normal growth control group, the Au NRs group, the rGO–COOH/Au NRs group, the DOX_rGO–COOH group, and the DOX_rGO–COOH/Au NRs group. Each group could be further divided into two parts, one part received no radiation and the other received NIR laser radiation (0.7 W cm^-2, 10 minutes). After laser treatment, tumor volume and mouse body weight for all the groups were measured every 2 days until 16 days. The tumor size was determined using a caliper and was calculated with the formula: tumor volume = (length × width^2)/2. **P < 0.01, *P < 0.05. All data were expressed as mean ± SD.

Results and discussion
Preparation and characterization of DOX_rGO–COOH/Au NRs multilayer
Firstly, the formed rGO–COOH was characterized by FTIR. As shown in Figure 1A, in the spectrum of GO, the broad and intense band at 3,406 cm^-1 can be assigned to the O–H stretching vibration. The peak at 1,730 cm^-1 can be attributed to the stretching vibration of C=O in carboxylic acid or carbonyl moieties, while the peak at 1,622 cm^-1 corresponds to C=C bonding of aromatic rings in the GO carbon skeleton structure. In the case of rGO–COOH, the C=O stretching vibration at 1,050 cm^-1 in the primary alcohol and the O–H stretching vibration at 3,400 cm^-1 significantly decrease. Meanwhile, the C=O stretching vibration at 1,221 cm^-1 in alkyl aryl ether disappear totally. These results confirmed the formation of rGO–COOH.

Then, DOX, a common anticancer drug, was adsorbed onto rGO–COOH by mixing the nanosheets with DOX solution, this absorption was based on π–π stacking and the electrostatic interaction between them. After the resulting DOX_rGO–COOH was collected by centrifugation, no free DOX could be detected in the supernatant by UV-Vis spectrometer, meaning that the loading percentage of DOX on rGO–COOH can be calculated as 100% according to the equation (1). It could also be expressed as 0.21 mg/mg of DOX adsorbed by rGO–COOH. Simultaneously, from the UV-visible spectra (Figure 1B), the peak at 488 nm can be observed in both DOX and DOX_rGO–COOH, which corresponds to the characteristic absorption of DOX.

Another nanomaterial, Au NRs, was fabricated by the seed-mediated method using CTAB as a surfactant, and the fabrication of Au NRs was established by UV-visible spectroscopy (Figure 1B). The absorption spectrum of Au NRs shows two peaks at 510 nm (radial surface plasma resonance) and 853 nm (longitudinal surface plasma resonance [LSPR]). From the TEM images (Figure 1C), the prepared Au NRs are observed to be well-dispersed and stable with 72.2 nm length and 16.6 nm width. While in Figure 1D, the lattice parameter of 0.235 nm in the high resolution TEM (HR-TEM) of Au NRs could correspond to Au 111 lattice. Meanwhile, the zeta potentials of Au NRs and rGO–COOH (Figure S1) were determined to be 42.4 ± 4.1 mV and -51.4 ± 4.5 mV, respectively, which facilitates the following assembly.
The as-developed DOX_rGO–COOH and Au NRs were assembled alternately via the LbL technique. From the UV-visible spectra (Figure 1B), characteristic absorptions of DOX at 488 nm and the LSPR peak of Au NRs at 858 nm are observed. Additionally, surface morphology of the multilayer is shown in Figure S2, and the presence of rGO nanosheets and Au NRs can be easily observed from the image. Meanwhile, the exterior appearance of the plaster was photographed and compared with blank nonwoven fabric (Figure S3). It is obvious that nonwoven textile could be used as a substrate to load DOX_rGO–COOH and Au NRs.

**DOX release behavior of DOX_rGO–COOH/Au NRs multilayer**

The drug release behavior of the multilayer was studied in PBS at 37°C, especially the effect of NIR radiation on the multilayer, as receiving or not receiving laser radiation (808 nm, 0.7 W cm⁻², 10 minutes). According to the equation (2), the cumulative release of DOX is profiled against time as shown in Figure 2, indicating that DOX is significantly released from the DOX_rGO–COOH/Au NRs multilayer when laser radiation is applied, while the release becomes obviously slow at laser-off state. The same experiment was performed on DOX_rGO–COOH and a similar release tendency is noted, except for a lower cumulative release of DOX. Thus, the NIR radiation apparently promoted the release of DOX at different time points, which may be attributed to the increase of mobility in DOX and the weakening of interaction between DOX and rGO–COOH resulting from the heat converted from NIR radiation. Moreover, the DOX release can be repeatedly activated by NIR light and modulated by the number of laser On/Off cycles. Therefore, the PT effect.

**Figure 1** (A) FTIR spectra of GO and rGO–COOH; (B) UV-Vis spectra of Au NRs, DOX, DOX_rGO–COOH, DOX_rGO–COOH/Au NRs, and rGO–COOH; (C) TEM image of Au NRs. The scale bar is 100 nm; (D) high-resolution TEM image of Au NRs. **Abbreviations:** Au NRs, gold nanorods; DOX, doxorubicin; FTIR, Fourier transform infrared spectroscopy; GO, graphene oxide; rGO–COOH, carboxylated-reduced GO; TEM, transmission electron microscope; UV-Vis, ultraviolet–visible spectrophotometry.
of the DOX_rGO–COOH/Au NRs multilayer repeatedly facilitated precise and controlled release of anticancer drugs, by which a large cumulative release of DOX (63.2%) was finally obtained. As a comparison, the cumulative release of DOX from the DOX_rGO–COOH/Au NRs multilayer with different layer number was investigated in the same way. As shown in Figure 2B, there is no significant difference among the studied systems, but the value for (DOX_rGO–COOH/Au NRs) with four bilayers is slightly lower than that of (DOX_rGO–COOH/Au NRs)4, meaning that the promoted release of DOX from the multilayer by laser illumination might be blocked somewhat by more layers than four bilayers.

**PT effects of rGO–COOH/Au NRs multilayer**

Depending on the strong NIR absorption of the rGO–COOH/Au NRs multilayer, a significant PT effect can be expectedly observed during NIR radiation. To investigate this effect, NIR laser (808 nm, 0.7 W cm⁻², 10 minutes) was applied to rGO–COOH dispersion in water (20 µg/mL), Au NRs dispersion in water (22 µg/mL), and (rGO–COOH/Au NRs)₄ multilayer with a diameter of 2 mm in water, respectively. Here, rGO–COOH or Au NRs was used in the same amount as that in the rGO–COOH/Au NRs multilayer. The change in temperature was recorded every 120 seconds. The ultrapure water was used as a control. Temperature trends of the four systems with the radiation time are shown in Figure 3A. For the Au NRs system, the temperature rises rapidly from 25°C to 49.8°C within 10 minutes. A rise in temperature from 25°C to 42.6°C is observed for rGO–COOH system, while the temperature of the pure water increases only about 2°C under similar conditions. However, during the radiation process, rGO–COOH/Au NRs significantly improve the PT heating effect as a temperature of 60.0°C is achieved, which may be attributed to the coupling of the PT effects of rGO and Au NRs. Furthermore, the rGO–COOH/Au NRs multilayer shows a photothermal conversion efficacy which is dependent on layer number (Figure 3B). The temperature rises with increasing layer number but from (rGO–COOH/Au NRs)₂ to (rGO–COOH/Au NRs)₄, the increase in temperature is not so significant as that from (rGO–COOH/Au NRs)₂ to (rGO–COOH/Au NRs)₄ and from (rGO–COOH/Au NRs)₂ to (rGO–COOH/Au NRs)₄. Similar as the above discussion of the DOX release, the photothermal conversion might be blocked by upper layers. Overall, these results suggest that the as-assembled rGO–COOH/Au NRs could be a rather effective PT converter of 808 nm laser radiation.

**Cytotoxicity of (DOX_rGO–COOH/Au NRs)₄ multilayer**

Owing to the excellent PT effects of the film, its therapeutic effect on cancer cells was evaluated. Seven experimental groups including TCPS control, blank nonwoven substrate, rGO–COOH, Au NRs, rGO–COOH/Au NRs, DOX_rGO–COOH, and DOX_rGO–COOH/Au NRs were maintained. Firstly, the cytotoxicity of all the groups in A-431 cells was investigated by CCK-8 assay (Figure 4A). After the cells are incubated with each group for 24 hours, there is no obvious cytotoxicity for TCPS, blank nonwoven fabrics, rGO–COOH, Au NRs, and rGO–COOH/Au NRs, where the cell viabilities are all higher than 90%, except for those of DOX_rGO–COOH and DOX_rGO–COOH/Au NRs, the cell
viabilities are much lower, 63% and 59%, respectively. This might be due to the released DOX from the film. Then, to further determine the PT therapeutic efficacy, the cells cultured for 12 hours in all the groups were exposed to 808 nm light at the power intensity of 0.7 W cm$^{-2}$ for 10 minutes followed by the cell viability assessment again. The results show that DOX_rGO–COOH/Au NRs multilayer can effectively kill the cells under NIR radiation showing the lowest cell viability of 17%, which suggests that the combinatorial therapy has a much higher efficacy than that of individual treatments. Additionally, the cells cultured for another 6 hours after laser irradiation were co-stained by Calcine AM (Ca AM) and PI.
to confirm live (green) and dead (red) cells, respectively. The results were consistent with the above quantitative assessment. In the confocal microscopy images (Figure 4B), cells in the control group display highly intense green fluorescence before and after NIR radiation, suggesting that the light does not damage the cells. Some cells incubated with rGO–COOH or Au NRs are dead under NIR radiation showing red fluorescence. This proves that rGO–COOH or Au NRs is cytotoxic by absorbing and converting the light to heat. Meanwhile, cytotoxicity is much higher for rGO–COOH/Au NRs under the same conditions. This can be attributed to the coupling of the SPR of Au NRs with the laser reduction of rGO–COOH. According to previous reports, we think that Au nanostructure can serve as photothermal sources through nonradiative decay and also act as local nanohotspot to enhance the optical energy absorption of graphene at a selected plasmon frequency. The most significant cell death is observed in the system with DOX_rGO–COOH/Au NRs with negligible green fluorescence. Taken together, the synergistic effects of PTT and CT are responsible for the improved therapeutic efficacy of the film.

In vivo evaluation of (DOX_rGO–COOH/Au NRs)₄ multilayer

This study aimed to demonstrate the application of the film for treating superficial tumors on the skin. With the positive in vitro results, we investigated the in vivo chemo-photothermal therapeutic effect in mice bearing human subcutaneous tumor cells, A-431. The as-developed DOX_rGO–COOH/Au NRs plaster was pasted onto the skin surface of the tumor sites at the beginning of the animal experiments. This treatment was nonsurgical and noninvasive.

The tumor-bearing mice were divided into seven groups, namely those with blank nonwoven substrate, rGO–COOH, Au NRs, rGO–COOH/Au NRs, DOX_rGO–COOH, DOX_rGO–COOH/Au NRs, and that without any treatment. Half of the mice in each group was subjected to NIR laser (808 nm, 0.7 W cm⁻²) for 10 minutes, and the remaining left half without radiation exposure were used as controls.

The antitumor therapeutic effect can be analyzed quantitatively by monitoring the change in tumor volume over time after treatment, and the tumor growth profiles are shown in Figure 5A. The growth rates of the seven groups are compared and a significant difference is noted. Tumors in the mice of the control groups exhibit continuous and rapid growth, even after NIR radiation, which indicates that the NIR laser does not have any damage to the tumors. In contrast, in the case of DOX_rGO–COOH-treated group, although initially there is slight inhibition of tumor growth without laser radiation due to the toxicity of DOX, the tumor finally grows to the same level as that in the group without any treatment; however, the tumor growth is suppressed a lot

![Figure 5 (Continued)](https://example.com/fig5cont)
after radiation in the same group. This suggests an enhanced toxicity of DOX due to its photo-activated release. In the case of Au NRs, the tumor grew rapidly without radiation, but with radiation, a strong tumor regression is observed. An almost similar tendency is noted for the rGO–COOH treated mice. But, rGO–COOH/Au NRs shows a better antitumor effect than rGO–COOH or Au NRs alone, such that the tumor does not grow further and was slightly suppressed. These data indicate that the combination of rGO–COOH and Au NRs shows high efficiency in converting NIR laser to heat energy, which can kill the tumor cells. Notably, DOX_rGO–COOH/Au NRs has the best antitumor effect among all the studied groups. For example, the tumor in the DOX_rGO–COOH/Au NRs-treated mice under NIR radiation is eliminated completely without recurrence. Moreover, the DOX_rGO–COOH/Au NRs-treated mice were quite healthy without toxic effects and tumor recurrence over 60 days, while those in the other six groups showed an average life span of no more than 30 days. The body weight of the mice in all the groups was measured and the comparison is shown in Figure 5B. There were no significant changes in body weight of all mice indicating a negligible toxicity of all the systems in this study. Additionally, the images of the DOX_rGO–COOH/Au NRs-treated mice carrying tumors under NIR radiation were acquired during the treatment (Figure 5C). From the images, it can be observed that the treatment leads to the appearance

**Figure 5** (A) The relative tumor size of mice in different treatment groups with or without NIR radiation. (B) The change in mouse body weight ratios in different treatment with or without NIR radiation. (C) Representative photographs of the mice at different times after treatment with DOX_rGO–COOH/Au NRs under NIR radiation. 

**Abbreviations:** Au NRs, gold nanorods; DOX, doxorubicin; rGO–COOH, carboxylated-reduced graphene oxide; NIR, near infrared.

**Figure 6** Images of H&E-stained tumor sections collected from the mice bearing tumors after various treatments including DOX_rGO–COOH + NIR radiation, rGO–COOH/Au NRs + NIR radiation, and DOX_rGO–COOH/Au NRs + NIR radiation. The tumors were also collected from the two control groups and the group only receiving NIR radiation. 

**Abbreviations:** Au NRs, gold nanorods; DOX, doxorubicin; rGO–COOH, carboxylated-reduced graphene oxide; NIR, near infrared.
of superficial scabs at tumor sites and the scabs fall off at day 8 without recurrence in the following days.

Furthermore, the tumor sections were subjected to histological analysis. H&E-stained tumor slices (Figure 6) reveal the presence of tumor lesions in the two control groups, and even under NIR radiation the lesions do not show any changes. In the case of the DOX_rGO–COOH-treated tumors receiving NIR radiation, the cells are destroyed partly. Obvious remissions of tumor cells can be observed only in rGO–COOH/Au NRs- or DOX_rGO–COOH/Au NRs-treated mice with NIR radiation. Taken together, these results suggest that the chemo-thermal synergistic therapy of the DOX_rGO–COOH/Au NRs plaster is effective and might be applicable for treating superficial tumors.

Conclusion
In this study, we fabricated a composite nanomedicine system, DOX_rGO–COOH/Au NRs, with nonwoven fabrics as the substrate. The system was developed to be used as an anticancer plaster, especially, for treatment of superficial tumors. The plaster was demonstrated to be a powerful PT agent. It produced heat faster than that produced by rGO–COOH or Au NRs alone when exposed to NIR radiation due to the synergistic effects between these two nanomaterials. Moreover, it also managed to enhance the release of encapsulated DOX when activated by NIR radiation. Accordingly, DOX_rGO–COOH/Au NRs showed high cytotoxicity and caused severe cell necrosis. Furthermore, the DOX_rGO–COOH/Au NRs plaster completely eradicated human squamous cell carcinoma in mice within 2 weeks without recurrence. Overall, the study suggests that the DOX_rGO–COOH/Au NRs plaster could be an effective, convenient, and noninvasive treatment option for treating superficial tumors.

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Disclosure
The authors report no conflicts of interest in this work.

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Supplementary materials

**Figure S1** Zeta-potential of Au NRs and rGO–COOH.
Abbreviations: Au NRs, gold nanorods; rGO–COOH, carboxylated-reduced graphene oxide.

**Figure S2** (A) AFM image of the surface of DOX_rGO–COOH/Au NRs multilayer. (B) The photo of DOX_rGO–COOH/Au NRs multilayer on nonwoven fabric.
Abbreviations: AFM, atomic force microscopy; Au NRs, gold nanorods; DOX, doxorubicin; rGO–COOH, carboxylated-reduced graphene oxide.

**Figure S3** The photos of nonwoven fabric with DOX_rGO–COOH/Au NRs multilayer. (A) blank nonwoven fabric; (B) (DOX_rGO–COOH/Au NRs); (C) (DOX_rGO–COOH/Au NRs).
Abbreviations: Au NRs, gold nanorods; DOX, doxorubicin; rGO–COOH, carboxylated-reduced graphene oxide.