A family AA5_2 carbohydrate oxidase from *Penicillium rubens* displays functional overlap across the AA5 family

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Abstract

Copper radical alcohol oxidases belonging to auxiliary activity family 5, subfamily 2 (AA5_2) catalyze the oxidation of galactose and galactosides, as well as aliphatic alcohols. Despite their broad applied potential, so far very few AA5_2 members have been biochemically characterized. We report the recombinant production and biochemical characterization of an AA5_2 oxidase from *Penicillium rubens Wisconsin 54–1255* (**Pru**AA5_2**A**), which groups within an unmapped clade phylogenetically distant from those comprising AA5_2 members characterized to date. **Pru**AA5_2 preferentially oxidized raffinose over galactose; however, its catalytic efficiency was 6.5 times higher on glycolaldehyde dimer compared to raffinose. Deep sequence analysis of characterized AA5_2 members highlighted amino acid pairs correlated to substrate range and conserved within the family. Moreover, **Pru**AA5_2 activity spans substrate preferences previously reported for AA5 subfamily 1 and 2 members, identifying possible functional overlap across the AA5 family.

Introduction

Biocatalysts developed to date to bolster the utilization of plant biomass have focused on the deconstruction of lignocellulose to sugars that can then be converted to fuels and chemicals [1,2]. On the other hand, the functional derivatization of plant material to make high-value bioproducts is a new area of biomass utilization research. Auxiliary Activity family 5 (AA5) members are copper radical oxidases (CROs) which are attractive targets for this purpose because of their ability to perform oxidation in a chemo-selective manner using only an inexpensive copper ion cofactor and oxygen. The AA5 family includes two subfamilies, namely AA5_1 and AA5_2, comprising characterized glyoxal oxidase and alcohol/carbohydrate oxidase enzymes respectively. So far, only few members of the AA5_2 subfamily (E.C.: 1.1.3.9) have been characterized, including the archetypal galactose oxidase from *Fusarium graminearum* (**Fgr**GaOx). Such galactose oxidases comprise an N-terminal carbohydrate-binding
module (CBM32, PF00754), a central catalytic domain containing three of the four copper-ligands (pfam Kelch_1 domain, PF01344), and a C-terminal domain (pfam DUF_1929 domain, PF09118) that provides the fourth copper ligand [3]. They catalyze the two-electron oxidation of C6-OH of D-galactose, generating the corresponding aldehyde while reducing molecular oxygen to hydrogen peroxide [4,5]. The aldehyde product can also be further oxidized to the carboxylic acid through oxidation of the geminal diol derivative of the aldehyde product [6]. While the $k_{\text{cat}}$ of FgrGaOx is approximately 100 times higher on D-galactose than galactose-containing polysaccharides, FgrGaOx shows nearly two times higher catalytic efficiency ($k_{\text{cat}}/K_m$) on galactoglucomannan and galactoxyloglucan compared to galactose [7]. The performance of FgrGaOx on galactose-containing polysaccharides has prompted its use in a broad range of applications [8], including hydrogels and aerogels [9–12], as well as cellulose coatings [13–15].

Previous work in our groups unveiled catalytic diversity within the AA5_2 subfamily beyond the galactose oxidases from Fusarium sp. Specifically, two AA5_2 homologs from the phytopathogenic fungi Colletotrichum graminicola (CgrAlcOx), and C. gloeosporioides (CglAlcOx) were characterized as general alcohol oxidases based on their high enzymatic activity towards aromatic and aliphatic alcohols, rather than carbohydrates [16]. Later, the raffinose oxidase from C. graminicola (CgrRaOx), containing a PAN_1 domain (PF00024) instead of the N-terminal CBM32 (PF0754) of FgrGaOx, was also reported [6]. In the present study we further investigated the protein sequence space within the AA5_2 subfamily using the catalytic modules from the CAZy database and sequence-function correlations of characterized AA5_2 members. Our analyses led to the selection of PruAA5_2A from Penicillium rubens Wisconsin 54–1255 (strain ATCC28089, UniprotKB: B6HHT0), which displayed dual activity preference on glycolaldehyde dimer and galactose-containing oligosaccharides, consistent with diverse biological functions.

Materials and methods

Chemicals and enzymes

Wild-type galactose oxidase from Fusarium graminearum was produced in Pichia pastoris KM71H and purified as previously described [17]. Horseradish peroxidase (P8375) and catalase from bovine liver (C40) were purchased from Sigma. If not otherwise specified, all chemicals and carbohydrate substrates were purchased from Sigma-Aldrich (USA). Galactoxyloglucan from tamarind was purchased from Megazyme (Ireland).

Sequence analyses

Fifty-two amino-acid sequences of fungal AA5_2 members and five amino-acid sequences of fungal AA5_1, corresponding to characterized members in the literature, were extracted from the public version of the CAZy database (http://www.cazy.org/AA5.html) [18]. In addition, homologs of CAP96757 were retrieved from the JGI Mycocosm portal [19] by blasting its full length sequence onto the Ascomycota. A total of three sequences with a percentage of identity superior to 60% were included in the analysis. Where present, signal peptides and additional modules, such as carbohydrate-binding modules, were removed to isolate the catalytic modules for subsequent analyses. A multiple sequence alignment was performed using MUSCLE [20] and a maximum likelihood phylogenetic tree was produced using RAxML v.8, with a 100 bootstrap, located on The CIPRES Science Gateway portal [21] (www.phylo.org). Subfamilies were inferred based on their bootstrap values (>75) and the tree was formatted using Figtree.

To identify amino acid positions likely to contribute to substrate range, an alignment was also performed for functionally characterized AA5_2 members, including the alcohol oxidase...
(AlcOx) (GenBank: EFQ30446.1) and raffinose oxidase (RaOx) (GenBank: EFQ36699) from \textit{Colletotrichum graminicola M1.001} (Table 1). In addition, Phyre2 was used to generate structure predictions for PruAA5\_2A and CgrRaOx based on homology-modeling [22]. Amino acid differences were mapped to the models and the crystal structures of FgrGaOx (PDBID 1GOG) [3] and CgrAlcOx (PDBID 5C92) [16]. They were then grouped according to function, where Group 1 included catalytic residues and copper-ligands, Group 2 include amino acid positions implicated in substrate range, and Group 3 included amino acid positions identified through mutagenesis to increase catalytic activity or stability. All figures were prepared with UCSF Chimera.

Based on the sequence analyses, PruAA5\_2A from \textit{Penicillium rubens} (strain ATCC 28089 / DSM 1075 / NRRL 1951 / Wisconsin 54–1255; Uniprot: B6HHT0; GenBank: 96757.1) was selected for recombinant protein production and characterization.

**Gene synthesis, cloning, expression and purification of PruAA5\_2A**

Prior to gene synthesis, the native signal sequence of PruAA5\_2A was predicted using the SignalIP server [23] and removed from the amino acid sequence. The gene encoding the resulting protein sequence, including prosequence, was optimized for expression in \textit{P. pastoris} and then synthesized and cloned into pPICZalpha by Genscript (NJ, USA). Selection of the \textit{P. pastoris} transformants and bioreactor expression and purification of PruAA5\_2A were performed as previously described [17]. Briefly, the expression vector encoding PruAA5\_2A was transformed into \textit{P. pastoris} SMD1168H by electroporation [24]. Transformants were then selected on YPD agar plates containing Zeocin (100 \mu g/ml) and screened for protein expression through colony blotting; supernatant samples from small scale cultivations (5 mL) were also screened for galactose and raffinose oxidase activity as described below. The transformant showing highest expression of PruAA5\_2A was then selected for PruAA5\_2A production in a bioreactor system following Invitrogen’s Pichia Fermentation Guidelines with minor modifications [17].

To purify PruAA5\_2A, the supernatant recovered from the bioreactor cultivation was adjusted to 1.5 M ammonium sulfate (pH 7.5) and then loaded on to a 20 mL sephacryl-phenol (high substitution) column (GE Lifesciences). Fractions containing PruAA5\_2A were then pooled and further purified by affinity chromatography using a 5 mL Ni-NTA column equilibrated with 50 mM sodium-phosphate buffer (pH 7.5) containing 20 mM imidazole and 500 mM NaCl. PruAA5\_2A was eluted from the Ni-NTA column by gradually increasing the imidazole concentration from 20 to 500 mM. Fractions containing purified PruAA5\_2A were then pooled, and the protein was concentrated and exchanged to 50 mM sodium phosphate (pH 7.5) using a Vivaspin 20 ultracentrifugation unit with a 30,000 MWCO cut-off (Satorius, Germany). The purity and molecular mass of PruAA5\_2A were assessed by SDS-PAGE using a gel imaging system and Image Lab software from Bio-Rad laboratories (USA). The protein concentration was determined using the Bradford protein assay from Bio-Rad Laboratories (USA) [25]. The final solution of PruAA5\_2A (3.5 mg/mL) was aliquoted in 50 \mu L fractions and submerged in liquid nitrogen for rapid freezing and stored then at -80˚C.

**Activity assay, substrate range and kinetics**

PruAA5\_2A activity was measured by following the formation of hydrogen peroxide using the previously described chromogenic ABTS (2,2’-azino-bis(3-ethylbenzothiazoline-6-sulphonic acid) and horseradish peroxidase (HRP) assay [26]. The final reaction mixture (volume: 205 \mu L) contained 7 U/mL horseradish peroxidase, 2 mM ABTS, and between 50 and 300 mM substrate in 20 mM MOPS buffer (pH 7.5), as the enzyme showed best performance at this
Prior to initiating the reaction, 5 μL of PruAA5_2A (40 ng) was incubated for 30 min at 30˚C in 100 μL 2x assay mix (4 mM ABTS and 15 U/mL HRP in milliQ water) to ensure complete activation of PruAA5_2A by HRP. The reaction was initiated by addition of a 2x substrate concentration (in 40 mM HEPES at pH 7.5) and continuously monitored for up to 3 h by reading the absorbance at 420 nm. Hydrogen peroxide production was calculated using the extinction coefficient of the ABTS radical as described in [6].

The substrate range of PruAA5_2A was determined using 300 mM D-glucose, L-arabinose, D-xylose, D-galactose, melibiose, sucrose, lactose, raffinose, stachyose, ethanol, 1-propanol, 2-propanol, 1-butanol, 1,2-butandiol, glycerol, D-sorbitol, benzyl alcohol; the exception was for the glycolaldehyde dimer, acetaldehyde, D-glyceraldehyde and glyoxalic acid where the activity was tested at 50, 25, and 15 mM of freshly prepared substrate solutions.

Kinetics parameters of PruAA5_2A and FgrGaOx were determined using 10 mM to 1600 mM for galactose and glycerol, 10 mM to 400 mM for raffinose and 10 mM to 150 mM for glycolaldehyde dimer using the activity assay as described above. Kinetic parameters were calculated using the Michealis-Menten function in Origin Pro 2016 (OriginLab Corp., USA), with the exception of the glycolaldehyde dimer where the substrate inhibition function was used instead of the Michaelis-Menten function.

Impact of pH, buffer and temperature. Effect of temperature on PruAA5_2A activity was determined by performing the activity assay described above with 300 mM galactose at 25, 30, 35, 40, 50 and 60˚C. The pH optimum was determined by performing the activity assay using 20 mM MOPS (pH 6.0 to 8.0), 20 mM HEPES (pH 6.0 to 8.5), or 25 mM sodium phosphate buffer (pH 6.0 to 8.0). To evaluate the impact of buffer type on the activity of PruAA5_2A,

#### Table 1. Amino acids and positions within characterized AA5_2 sequences that are implicated in catalysis and substrate preference.

| Category                  | Position | Amino acid | Reported function                                                                 | Reference(s) | CgrRaOx | CgrAlcOx | PruAA5_2 |
|---------------------------|----------|------------|-----------------------------------------------------------------------------------|--------------|---------|----------|----------|
| Catalysis                 | 194      | F          | \(\pi-\pi\) interaction with F227 and W290                                        | [39]         | F       | W        | Y        |
|                           | 228      | C          | C228-Y272 redox cofactor                                                          | [3,5,40]     | C       | C        | C        |
|                           | 272      | Y          | C228-Y272 redox cofactor, copper ligand                                           | [3,5,40]     | Y       | Y        | Y        |
|                           | 441      | F          | \(\pi-\pi\) stacking interaction with Y495                                        | [3]          | F       | F        | F        |
|                           | 464      | F          | \(\pi-\pi\) stacking interaction with Y495                                        | [39]         | F       | F        | F        |
|                           | 495      | Y          | Catalytic tyrosine, copper ligand                                                 | [3,5]        | Y       | Y        | Y        |
|                           | 496      | H          | Copper ligand                                                                     | [3,5]        | H       | H        | H        |
|                           | 581      | H          | Copper ligand                                                                     | [3,5]        | H       | H        | H        |
| Substrate preference      | 290      | W          | \(\pi-\pi\) stacking interaction with Y272 and F194, hydrogen bonding with D-galactose | [3,5,16,34,40–42] | Y       | F        | W        |
|                           | 326      | Q\(^a\)    | Hydrogen bonding with R330                                                         | [34,41]      | A       | G        | D        |
|                           | 329      | Y          | Hydrogen bonding with D-galactose and R330                                        | [34]         | W       | M        | Y        |
|                           | 330      | R          | Bidentate hydrogen bonds with D-galactose                                        | [34,37,41]   | R       | F        | R        |
|                           | 405      | Y          | Hydrogen bonding with Y495                                                         | [32]         | Y       | Y        | Y        |
|                           | 406      | Q\(^a\)    | Hydrogen bonding with D-galactose                                                 | [34,37]      | S       | T        | E        |
| Catalytic efficiency      | 383      | C\(^a\)    | C383S increases \(V_{max}\) on D-galactose by 1.75x and lowers \(K_M\) by 3.6x     | [35,36]      | C       | C        | S        |
|                           | 436      | Y\(^a\)    | Y436H increases \(V_{max}\) on D-galactose by 2x                                  | [35]         | K       | Y        | A        |
|                           | 494      | V\(^a\)    | V494A increases \(V_{max}\) on D-galactose by 1.75x                              | [35,36,41]   | N       | N        | V        |

Amino acid positions correspond to the archetypal galactose oxidase from *Fusarium graminearum*.

\(^a\)Amino acids not fully conserved among AA5_2 sequences from other Fusarium species (Fig A in S1 File).

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Activity assays were also performed using following buffers and buffer concentrations: sodium phosphate (10, 25, 50 and 100 mM), potassium phosphate (10, 25 and 50 mM), HEPES (20, 40, 100 mM), MOPS (20, 40, 100 and 200 mM) and Tris-HCl buffer (20 mM) at pH 7.5.

Lag-phase analysis from activity data. The Gen5 microplate-reader software (BioTek, USA) was used to evaluate the lag-phase behavior of PruAA5_2 as a function of buffer type and pH value, where the kinetic lag-time is time defined as the interval between the line of the inception point (maximum slope) and the absorbance baseline.

Identification of oxidation products by MS and NMR
The negative and positive ionization MS and MS/MS spectra were produced using an Agilent 1100 Series LC/MSD Trap SL (Agilent Technologies Inc., Palo Alto, CA, USA) combined with electrospray ionization source. PruAA5_2A oxidation of raffinose was performed in 200 μL sterile water with 25 mM raffinose (12.6 mg/mL), 1 U/mg raffinose of horse radish peroxide, 115 U/mg raffinose of catalase, and 1 U/mg raffinose of PruAA5_2A or FgrGaOx. The specific activity of PruAA5_2A and FgrGaOx used in these reactions was 33 U/mg and 161 U/mg, respectively. Reactions were shaken at 600 rpm for 48 hours at 30°C; samples were then diluted (1/200) in 50% methanol and 1% formic acid. To form chloride adducts in negative mode, 0.5 μL ammonium chloride was added. Samples were directly infused at 5 μL/min, and ionization parameters were as follows: drying gas 4 L/min, nebulizer pressure 10 psi, temperature 325°C and capillary voltage 3500 V.

PruAA5_2A oxidation of sucrose was performed similar as described above but using 50 mM sucrose and 2 U/mg sucrose of PruAA5_2A. The specific activity of PruAA5_2A was 4.8 U/mg on sucrose. A sample containing the oxidation products of sucrose (2 mg) was analyzed by nuclear magnetic resonance (NMR) spectrometry. NMR measurements were performed at the ¹H frequency of 850 MHz (sample in D₂O) and 600 MHz (sample in DMSO) on Bruker Advance III HD spectrometers both equipped with a triple resonance cryogenic probes at 298 K.

Results
Comparative analysis of characterized and engineered AA5_2 sequences to inform sequence selection
A key aim of this study was to probe unexplored regions of the AA5_2 phylogeny to identify an AA5_2 member with a divergent substrate profile, which would further elucidate sequence-function relationships within this protein family. Accordingly, a phylogenetic tree was constructed which underscored the clear distinction between AA5_1 and AA5_2 members, and displayed 11 subgroups within the AA5_2 subfamily (bootstrap value >75) for which only four contain characterized members (Fig 1). PruAA5_2A from Penicillium rubens Wisconsin is included within the AA5_2 subfamily and clusters with homologs from other Penicillium species in a subgroup clearly separated from the rest of the other AA5_2 sequences.

Alignment of the current AA5_2 sequences from the Fusarium genus, including FgrGaOx, revealed 10 consensus regions in the catalytic domain of galactose oxidase (Fig A in S1 File). Corresponding consensus regions were also highly conserved within AA5_2 sequences from other organisms. For example, PruAA5_2A shares nearly 50% overall sequence identity to FgrGaOx, and 84% identity within consensus regions. Not surprisingly, amino acids that play a direct role in catalysis locate within the conserved sequence stretches, whereas amino acids contributing to substrate preference or catalytic efficiency mostly lie outside these regions (Table 1).
Closer inspection of the *PruAA5_2A* primary sequence identified an N-terminal F5_F8-type_C (CBM32) domain, a central 7-bladed β-propeller (Kelch_1 repeat) catalytic domain, and a C-terminal DUF1929 domain. *PruAA5_2A* includes the main galactose ligand arginine (Arg327), which corresponds to Arg330 in *FgrGaOx*. However, *PruAA5_2A* contains an aspartic acid (Asp326) at the position of Gln326 in *FgrGaOx*, which is involved in coordinating the position of Arg330 through hydrogen bonding [33,34]. *PruAA5_2A* also contains a serine that corresponds to the C383S substitution in *FgrGaOx* leading to nearly five times increased catalytic efficiency [35,36]. Considering amino acid positions believed to influence substrate binding, Gln406 of *FgrGaOx* that interacts with the C2-hydroxyl of galactose is instead glutamic acid (GaOx), which could potentially also impact substrate range [38].

Amino acids in *PruAA5_2A* corresponding to those listed in Table 1 were predicted through structural modeling using the Phyre2 server; 98% of the residues were modelled at >90% confidence (Fig 2A and 2B). The stereochemical quality of the predicted model was evaluated through RAMPAGE [43], for which the Ramachandran plot predicted 98% residues lying in the most favored region. This model was compared with the crystal structures of...
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Fgr GaOx (PDBID GOG1) and Cgr AlcOx (PDBID 5C92) and a model of the Cgr RaOx catalytic domain that covers 55% of the C-terminal end of the sequence with 100% confidence (Fig 2C–2F). The comparison of model and solved protein structures highlighted a region within the catalytic domain of characterized AA5_2 members that is consistently enriched in aromatic amino acids that are thought to play an important role in catalysis and stability of the copper-radical oxidase (Fig 2C; Table 1), along with the frequent substitution of residues on the opposing face that are correlated to substrate preference and catalytic performance (Fig 2D; Table 1).

The unique substitutions in Pru AA5_2A relative to previously characterized AA5_2 members suggested the enzyme would target galactose-containing carbohydrates but display a distinct substrate profile compared to archetypal galactose oxidases. Accordingly, Pru AA5_2A was selected for recombinant production and biochemical characterization.

Production of Pru AA5_2A

Bioreactor cultivation and downstream purification yielded 31 mg of PruAA5_2A per liter of cultivation with >90% purity (assessed by SDS-PAGE; Fig B in S1 File). Whereas the calculated molecular mass of PruAA5_2A is 70 kDa, the electrophoretic molecular mass of PruAA5_2A expressed in Pichia pastoris SMD1168H was 82 kDa, suggesting glycosylation of the protein. Indeed, 5 potential N-linked and 27 O-linked potential glycosylation sites were predicted in the PruAA5_2A sequence using the GlycoEP server [44]. Recombinant AA5_2 enzymes can comprise a mixture of those lacking copper or the Tyr-Cys thioether crosslink, and the mature oxidase (Cys228-Tyr227-Cu) [45]; accordingly, the purified PruAA5_2A was activated using 0.5 mM copper(II) sulfate as previously described [17,46], prior to characterization.

General biochemical properties

PruAA5_2A activity on galactose was optimal at pH 7.5 (Fig 3A) and the enzyme showed higher activity in HEPES or MOPS buffers compared to potassium phosphate, sodium phosphate, and Tris-Cl buffers. Specifically, PruAA5_2A activity dropped by 2.7 times when increasing sodium phosphate concentration from 25 to 100 mM (Fig 3B), suggesting that phosphate ions could inhibit substrate oxidation through unfavorable interaction with the copper(II)-ion in PruAA5_2A, as has been reported for other copper-containing oxidases (further discussed below). Also, PruAA5_2A showed lower activity in potassium phosphate and Tris-Cl buffers relative to the sodium phosphate buffer. The highest activity of PruAA5_2A was measured at 50°C; however, 70% of PruAA5_2A activity was lost after eight hours at this temperature (Fig C in S1 File). Given the noted impacts of buffer type and temperature on PruAA5_2A activity, all activity analyses were performed at 30°C in 20 mM MOPS (pH 7.5), unless otherwise mentioned.

For all tested conditions, PruAA5_2A exhibited an initial lag-phase as shown for reactions performed in MOPS and sodium phosphate buffers (Fig 4A). In phosphate buffer (pH 7.5), the
**Substrate profile of PruAA5_2A.** Of the carbohydrates tested, PruAA5_2A displayed highest activity towards galactopyranosyl-α-(1–6)-substituted oligosaccharides, including raffinose (31.4 U/mg), followed by melibiose (17.5 U/mg) and then stachyose (9.4 U/mg) (Fig 5). Activity on lactose was comparatively low (2.4 U/mg), pointing to the importance of the α-(1–6)-glycoside bond of the target galactopyranosyl unit. Of note, the preference of PruAA5_2A for galactopyranosyl-α-(1–6)-oligosaccharides is reminiscent of two recently discovered AA5_2 members from *Fusarium sambucinum*, FsaGaOx, [47] and *Colletotrichum graminicola*, CgrRaOx [6]. Moreover, despite higher activity on oligosaccharides over monosaccharides, PruAA5_2A was, like CgrRaOx, not active on galactose-containing polysaccharides, including galactoxyloglucan and galactoglucomannan (data not shown). Products generated by

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**Fig 3. Influence of pH and choice of buffer on PruAA5_2A activity.** (A) Activity as a function of pH established in 25 mM MOPS, 25 mM HEPES, or 25 mM sodium phosphate. (B) Influence of buffer type and concentration on PruAA5_2A activity on 150 mM raffinose (pH 7.5). n = 4; error bars indicate standard deviation.

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PruAA5_2A oxidation of raffinose were analyzed by MS to confirm oxidation of the C-6 hydroxyl of the galactosyl residue. Identical to reaction products generated by FgrGaOx and CgrRaOx (see Fig D in S1 File and [6]), the main product generated by PruAA5_2A was indeed oxidized at the C-6 of the galactosyl residue of raffinose (m/z 569). Like CgrRaOx, the further oxidation product, carboxylic acid, was also present in minor amount (m/z 517).

Surprisingly, clear activity was found on the disaccharide sucrose (4.8 U/mg) although no activity could be detected for D-glucose.

1H NMR analysis was therefore conducted in both D2O and DMSO-d6 to determine whether PruAA5_2A could target the fructosyl residue in sucrose [48]. Unfortunately, in both solvents, observed chemical shifts overlapped with the starting material, and thus were impossible to further analyze (Fig E in S1 File). Due to the low concentration of the product, no attempt to run 2D NMR was made.

Compounds other than carbohydrates were generally poor substrates; and similar to other AA5_2 members, PruAA5_2A did not oxidize the secondary alcohol 2-propanol. The
exception was for glycerol and freshly prepared solutions of glycolaldehyde dimer (50 mM), where the substrate concentration was 50 mM. Activity on Glyoxalic acid was measured at 15 mM since no activity was detected at 50 mM. In all cases, reactions were performed at 30°C in 20 mM MOPS (pH 7.5). n = 4; error bars indicate standard deviation.

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Fig 5. Substrate range of PruAA5_2A. PruAA5_2A activity was measured using 300 mM substrate, except for glyceraldehyde, acetaldehyde and glycolaldehyde dimer, where the substrate concentration was 50 mM. Activity on Glyoxalic acid was measured at 15 mM since no activity was detected at 50 mM. In all cases, reactions were performed at 30°C in 20 mM MOPS (pH 7.5). n = 4; error bars indicate standard deviation.

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PruAA5_2A activity was measured using 15 mM glyoxalic acid, which could be explained by the formation of the hydrate form (geminal diol) of the aldehyde group.

**Kinetic properties of PruAA5_2A.** Given the limited range of solubility for raffinose (500 mM) and non-saturated behavior of PruAA5_2A kinetic values reported herein are apparent values. Kinetic analyses using preferred substrates revealed that the apparent catalytic efficiency of PruAA5_2A was nearly 2.5 times higher on raffinose compared to galactose (Fig 6). By contrast, the catalytic efficiency of FgrGaOx on raffinose and galactose was similar (30 s⁻¹ mM⁻¹ and 20 s⁻¹ mM⁻¹ respectively, see Fig F in S1 File; enzyme production described in [17]). Whereas FgrGaOx activity on glycolaldehyde dimer has not been detected [6], the best kinetic performance of PruAA5_2A was observed using this substrate, where the apparent catalytic efficiency (k_{cat}/K_M) of PruAA5_2A on freshly prepared glycolaldehyde dimer was 6.5 times higher than raffinose. The corresponding kinetic profile,

![Kinetic analysis of PruAA5_2A on preferred substrates](https://doi.org/10.1371/journal.pone.0216546.g006)

Fig 6. Kinetic analysis of PruAA5_2A on preferred substrates. Raffinose (■), galactose (●), glycerol (▲) and a fresh solution of glycolaldehyde dimer (▼). n = 4; error bars indicate standard deviations. The data were fitted to the Michaelis-Menten or substrate-inhibition (glycolaldehyde dimer) functions using the OriginPro analysis software (iteration algorithm: Levenberg-Marquardt); in all cases R² values were > 0.95. For all substrates besides the glycolaldehyde dimer, saturation kinetics were not achieved below substrate solubility. Accordingly, apparent kinetic parameters are reported.

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however, was consistent with substrate inhibition by the glycolaldehyde dimer \((K_i = 178 \text{ mM})\). In solution, the dimer form rearranges into glycolaldehyde and the hydrate form (Fig 5) where the hydrate form is the major component (70%) at equilibrium [49]. Given the loss of activity during storage together with the apparent substrate inhibition of glycolaldehyde dimer, it is conceivable that the hydrate form, or other derivatives of glycolaldehyde, are inhibitors of PruAA5_2A activity.

**Discussion**

PruAA5_2A represents the first family AA2_2 member retrieved from the Eurotiomycetes order to be investigated; all other characterized fungal AA5_2 members are from *Fusarium* and *Colletotrichum* species, belonging to the Sordariomycetes order. This protein was present within a distinct group in our phylogenetic study, and comparisons between PruAA5_2A and other characterized fungal AA5_2 sequences highlighted amino acids that likely contribute to substrate preference (Table 1).

First, the amino acids at positions Phe194, Phe441, and Phe464 of *Fgr*GaOx are conserved among AA5_2 copper-radical oxidases and appear to play a role in stabilizing the radical electron or position of the Tyr495 copper ligand, thereby stabilizing the active form of the enzyme. These positions are in addition to Trp290, which contributes to pi-stacking interactions with the tyrosycysteine linkage, but is also suggested to act as a galactose ligand [40]. In this context, it is interesting to note that residues corresponding to Trp290 and Phe194 in both CgrAlcOx and CglAlcOx are swapped to phenylalanine and tryptophan, respectively (Fig 2E). By contrast, Trp290 is replaced by tyrosine and Tyr320 by tryptophan in CgrRaOx (Fig 2F), whereas Phe194 is replaced by a tyrosine in PruAA5_2A (Fig 2B). It is conceivable then, that a tyrosine at the edge of the active center facilitates hydrogen bonding to the glucopyranosyl unit that neighbors galactose in melibiose and raffinose, thereby increasing the selectivity of CgrRaOx and PruAA5_2A for these substrates. Considering the poor activity on lactose, the orientation of the glucopuranosyl also plays an important role in interaction, supporting the notion that these enzymes seem to specifically target raffinose.

Second, Arg330 and Tyr329 in *Fgr*GaOx appear critical for activity on galactose [3,34,41], and are present in PruAA5_2A; however, the presence of an aspartic acid at position Gln326 (in *Fgr*GaOx) and glutamic acid at position Gln406 (in *Fgr*GaOx) could further facilitate the binding of oligosaccharides or broaden the substrate acceptance [37] (Fig 2B). Notably, positions Gln326 and Gln406 in *Fgr*GaOx are alanine and serine in CgrRaOx, which might explain the comparatively poor catalytic turnover of CgrRaOx on galactose and raffinose, as well as lack of detectable activity on stachyose and lactose since this enzyme seemingly lacks two of the galactose ligands (Fig 2F) [6]. Also notable, Arg330 and Tyr329 in *Fgr*GaOx are replaced by phenylalanine and methionine in CgrAlcOx (Fig 2E). The new analyses of PruAA5_2A, together with previous characterizations of AA5_2 oxidases, begin to suggest that the Arg330-Tyr329 pair, as well as Phe194-Trp290 pair, may delineate carbohydrate versus alcohol oxidase functionally within this enzyme subfamily.

Distinct from *Fgr*GaOx but similar to CgrRaOx, PruAA5_2A displayed lower \(K_M\) and higher catalytic efficiency on raffinose than galactose, and appeared unable to oxidize galactose-containing polysaccharides. Both PruAA5_2A and CgrRaOx are further distinguished from *Fgr*GaOx by their oxidation of the glycolaldehyde dimer. While sequence attributes leading to these activity differences were difficult to predict, it is interesting to note that the kinetic efficiency of PruAA5_2A on the glycolaldehyde dimer was nearly seven times higher than tested carbohydrate substrates, where the corresponding \(K_M\) value (53 mM) is comparable to that of *Fgr*GaOx on galactose.
Lag-phases and buffer inhibition were not previously reported for other biochemically characterized AA5_2 oxidases; however, similar impacts of phosphate and Tris buffers have been observed for other metal-containing oxidoreductases. For example, inhibitory effects of phosphate and Tris buffers have been observed for iron-lipoxygenases [50,51], copper-tyrosinases [52] and copper and zinc dismutases [53]. Whereas PruAA5_2A is the first wild-type AA5_2 member reported to display this phenotype, a lag phase was observed for glucose-oxidizing mutants of FgrGaOx, which was ascribed to the W290F substitution and substrate-induced transformation of the W290F variant to the more active form [42]. The substrate dependence of the lag phase observed for PruAA5_2A is consistent with this model, but suggests it is not only attributed to the W290F substitution, given that PruAA5_2A contains a tryptophan at the corresponding position.

The biological functions of family AA5_2 oxidases remain elusive; however, conceivable options are beginning to materialize through the increasing number of AA5_2 members that display activity beyond galactose. For example, certain AA5_2 members may play a role in pathogenesis. Oxidation of glycolaldehyde may be a pathway for synthethis glyoxalic acid, which is implicated in fungal virulence [54]. Similarly, the activity of PruAA5_2A and CgrRaOx on raffinose points to a possible role in inhibiting oxidative stress responses in plants. Briefly, raffinose is a substrate of stachyose synthase [6,55], and stachyose along with verbascose are efficient oxygen radical scavengers in plants [56]; thus inhibition of their synthesis could weaken defense mechanisms during fungal attack. Likewise, oxidation of oligogalacturonides by AA5_2 enzymes could reduce the activation of plant immune responses during fungal attack [57].

The comparatively high activity of PruAA5_2A on the glycolaldehyde dimer solution is also reminiscent of the family AA5_1 CRO2 oxidase from Phanerochaete chrysosporium [58] and the copper-radical oxidase GlxA from Streptomyces lividans [59]. Whereas the biological role of CRO2 is unclear, family AA5_1 glyoxal oxidases (GLOXs) have already been implicated in the filamentous growth of phytopathogenic fungi [60]. GlxA does not belong to either AA5_1 or AA5_2 subclasses; however GlxA is a membrane-associated galactose-oxidase like cuproenzyme where the catalytic domain adopts a β-propeller fold and the C-terminus includes a DUF1929 domain [59]. Similar to PruAA5_2A, GlxA does not accept glyoxal, shows highest activity on glycolaldehyde ($K_M$ of 150 mM), and oxidizes galactose and glycerol albeit with $K_M$ values above tested substrate concentrations (i.e., above 0.6 M) [59]. While the natural substrate of GlxA is unclear, GlxA contributes to β-glycan synthesis and/or modification at hypal tips [59,61], impacting aerial growth, pellet formation, and response to osmotic stress [61–63]. Orthologues of GlxA are found throughout the Streptomyces genus and are believed to have been acquired through horizontal gene transfer from fungi [62]. Accordingly, a compelling possibility is that PruAA5_2A and other AA5_2 members likewise contribute to fungal cell wall remodeling.

To conclude, the dual activity of PruAA5_2A on both glycolaldehyde dimer as well as carbohydrates spans the substrate range reported for AA5_1, AA5_2, and unclassified AA5 oxidases. The diversity of low molecular weight substrates accepted by PruAA5_2A also reveals the potential of single AA5_2 members to contribute to multiple and diverse biological functions. The remaining difficulty in identifying the natural substrate of PruAA5 and other AA5_2 members, together with the occurrence of AA5_2 sequences predominately in fungal plant pathogens, suggests that at least some AA5_2 members act on low-abundant molecules involved in pathogeneticty and/or defence. Localizing AA5_2 activity in corresponding fungi during pathogenesis could shed light on the biological role and preferred substrates of this enzyme family.
Supporting information

S1 File. A new family AA5_2 member displays dual activity preference. Fig A. Analysis of AA5_2 Sequences. (A) Sequence conservation between FgrGaOx, PruAA5_2A, CgrRaOx and CgrAlcOx in within 10 sequence stretches recognized for having identical amino acids in the 9 analyzed Fusarium spp. Yellow highlights identical amino acids throughout the four sequences of FgrGaOx, PruAA5_2A, CgrRaOx and CgrAlcOx. Amino acids that also occur in Table 1 are in bold, red positions denotes variances. The placements of the sequence segments in the structure of FgrGaOx are highlighted in Fig 2C. (B) Sequence alignment of the 9 Fusarium spp. in subfamily AA5_2 used to identify the 10 conserved sequence segments and information presented in Table 1. The conserved sequence segments were defined as strings of 4 or more consecutive and conserved amino acids in alignment of the Fusarium spp. sequences.

Fig B. SDS-PAGE of pure PruAA5_2A after production and purification. The molecular weight of PruAA5_2A was estimated to be 75 kDa, indicating that the purified enzyme is glycosylated.

Fig C. Effect of temperature on PruAA5_2A activity. (A) Activity on 300 mM raffinose in 20 mM MOPS (pH 7.5). (B) Residual activity at 30˚C after 15, 30 min or 1, 2, 4 and 24 hours at 22, 30, 40, 50 and 60˚C in 20 mM MOPS (pH 7.5). n = 4; error bars indicate standard deviation.

Fig D. Analysis of oxidized products by mass spectrometry. (A) Negative mode ESI-MS spectra of oxidized raffinose produced by CgrRaOx -catalyzed reaction (3), and (B) by PruAA5_2A -catalyzed reaction. m/z 517, uronic acid; m/z 539, unoxidized raffinose (Cl adduct); m/z 569, aldehyde product reacted with methanol (Cl adduct); m/z 207.9, MOPS buffer.

Fig E. 1H NMR spectra of sucrose oxidation products. (A) Analyzed in D2O, and (B) DMSO-d6. Chemical shifts indicating the oxidized product formation and putative oxidation sites as hydrates (A) and aldehydes (B) marked with a ring. Due to low degree of oxidation the final structure of the oxidation product could not be determined.

Fig F. Comparative plot of PruAA5_2A and FgrGaOx substrate kinetics using the calculated Michealis-Menten plot from kinetical analysis. The activity axis (y-axis) were converted to relative activity for easy comparison. Actual kinetic parameters are present in Fig 6 for PruAA5_2A and the discussion for FgrGaOx. No activity on glycolaldehyde dimer was detected for FgrGaOx.

(PDF)

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