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Insights into HONO sources from observations during a solar eclipse

Ajit Singh¹, Leigh R. Crilley², Francis D. Pope¹ and William J. Bloss¹*

¹School of Geography, Earth and Environmental Sciences, University of Birmingham, Birmingham, B15 2TT, UK
²Department of Chemistry, York University, Toronto, Canada

*Corresponding author: William J. Bloss (W.J.Bloss@bham.ac.uk)

Abstract

Nitrous acid (HONO) is a major, and often the dominant, precursor to primary OH radical production in the daytime boundary layer, driving the removal of many primary pollutants and formation of secondary species such as ozone and many aerosol components. A number of photochemical HONO production mechanisms have been proposed, alongside homogeneous gas-phase reactions, to account for field observations of daytime HONO. The range of production mechanisms show varying dependencies upon precursor species such as NO₂, available surfaces for heterogeneous reactions, and dark / photoenhanced aspects. Here, we exploit measurements of HONO and related species during a near-total solar eclipse as a natural perturbation to the atmospheric photochemistry to assess the characteristics of the production mechanisms occurring at an urban background location. Little variation in HONO abundance was observed in response to changing light levels during the eclipse, pointing to relatively balanced photochemical source and (well-understood) sink terms. We employ a series of simple kinetic simulations to explore the consistency of different potential source mechanisms with the observations, finding evidence for a dominant role for photochemical processing of traffic-derived NO₂ upon surfaces producing HONO, alongside indications of a smaller contribution from direct vehicular emissions. Other mechanisms involving dark heterogeneous reactions were not, in isolation, consistent with the observations. The critical role of NO₂, ultimately derived overwhelmingly from local road traffic emissions at this location, points to significant future reductions in daytime HONO production with vehicle fleet evolution and reduction of tailpipe emissions.

Environmental Significance
Nitrous acid (HONO) is a major, often the dominant, boundary layer precursor to the key daytime atmospheric oxidant OH; however sources of HONO are poorly understood, with a number of candidate formation mechanisms advanced. Here, we use a natural perturbation – a solar eclipse – to identify the photochemical factors governing HONO formation, from their variation with this large-scale change in solar intensity. We demonstrate that HONO is not controlled by a single source, but that photoenhanced production (i.e. reactions accelerated by sunlight) are a major component of HONO production, suggesting a dominant role for photochemical processing of traffic-derived NO\(_2\) upon surfaces producing HONO, alongside a smaller contribution from direct vehicular emissions. In both cases, association with (anthropogenically derived) NO\(_2\) indicates that HONO formation may fall with future vehicle fleet evolution and tailpipe emissions reduction.

1 Introduction

Atmospheric chemical processing in the sunlit troposphere is driven primarily by the OH radical, which initiates the removal of most organic compounds, and drives the formation of ozone, and secondary organic and inorganic aerosols.\(^1\) While the dominant primary source of OH in the free troposphere is the photolysis of ozone and subsequent reaction of electronically excited oxygen atoms with water vapour, a series of recent field campaigns have shown that the photolysis of nitrous acid (HONO) is a major / the dominant primary OH precursor in the continental boundary layer.\(^3\)\(^,\)\(^4\)

\[
\text{HONO} + \text{hv} (\lambda < 440 \text{ nm}) \rightarrow \text{NO} + \text{OH} \quad \text{(R1)}
\]

HONO is formed from the slow homogeneous reaction between OH and NO, and in this sense acts as a photolabile reservoir for OH (and NO) with R1 and R2 forming a null cycle; however observed levels of HONO during daytime are typically one order of magnitude higher than R1 and R2 in isolation predict, indicating the presence of additional HONO formation mechanisms or emissions (and hence net OH production).\(^5\)\(^-\)\(^7\) The dark formation of HONO through the heterogeneous hydrolysis of nitrogen dioxide (R3 below) has been known for a number of years;\(^8\) however recent laboratory and field students have identified a number of additional candidate HONO formation mechanisms and/or emission sources.

\[
\text{OH} + \text{NO} + \text{M} \rightarrow \text{HONO} + \text{M} \quad \text{(R2)}
\]

\[
2\text{NO}_2 + \text{H}_2\text{O} \rightarrow \text{HONO} + \text{HNO}_3 \quad \text{(R3)}
\]
Primary sources of HONO include vehicle emissions\textsuperscript{9, 10}, biomass burning\textsuperscript{11-13}, microbial activities in soils\textsuperscript{14, 15} and associated with biocrusts\textsuperscript{16}. Laboratory studies have shown that photoenhanced HONO production can occur following NO\textsubscript{2} uptake to surfaces including soot\textsuperscript{17}, aromatic species\textsuperscript{18}, humic acids\textsuperscript{19, 20} and TiO\textsubscript{2}\textsuperscript{21}. Photoenhanced NO\textsubscript{2} heterogeneous reactions on ‘urban grime’ on building surfaces\textsuperscript{22} have also been shown to generate appreciable levels of HONO. Further potential sources are photolysis of nitric acid\textsuperscript{23}, nitrophenols\textsuperscript{24} and of particulate nitrate\textsuperscript{25-27}.

Many field and laboratory studies have been carried out to investigate the sources of HONO in different environments, e.g. urban, rural, coastal, forest and vehicle tunnels.\textsuperscript{5, 9, 28-34} In a recent study, Tong, et al.\textsuperscript{35} performed measurements at urban and suburban environments in Beijing during winter, and observed that direct emission and homogenous gas phase sources made a larger contribution in urban areas, while heterogeneous sources were suggested to be more significant away from urban centres. A recurrent challenge remains quantitatively reconciling daytime HONO concentrations in many urban environments with the current known sources.\textsuperscript{5-7}

Here, we take advantage of a near-total solar eclipse (~90% attenuation) as a natural short-term perturbation to atmospheric photochemistry to explore the chemical processes affecting HONO abundance. The reduction in photolysis frequencies during an eclipse is effectively uniform over a large area (relative to the chemical lifetime and hence spatial footprint of HONO, NO\textsubscript{x} and related species), of relevance to the challenges of heterogeneity.\textsuperscript{36, 37} Comparison of observed temporal behaviour during the eclipse therefore provides a unique test of our understanding of HONO photochemistry. We report the temporal variation of HONO, NO\textsubscript{x}, O\textsubscript{3} and aerosol characteristics during the eclipse, and use these measurements to explore the nature of sources of HONO which are consistent with their behaviour.

2 Observations during a solar eclipse

HONO, NO\textsubscript{x}, O\textsubscript{3} and particle number concentration were measured at an urban background monitoring station situated (52.45° N, 1.93° W) within the Edgbaston campus of the University of Birmingham, UK. Possible anthropogenic emission sources located nearby include a suburban rail line (north-west, ca. 90 m away), and a suburban road (east, 125 m away). HONO was measured using a Long-path Absorption Photometer (LOPAP).\textsuperscript{38} Briefly, the LOPAP is a wet chemical technique, where gas phase HONO is sampled, using a
stripping coil, into an acidic solution and is derivatived into an azo dye. The light absorption at 550 nm of the azo dye is then measured with a spectrometer using an optical path length of 2.4 m. The LOPAP was operated and calibrated according to the standard procedures, with a sampling height of 3m above ground level, and data acquired at 5 minute time resolution. Baseline (zero) measurements, obtained by sampling zero air, were taken at frequent intervals (8 hours). The detection limit (2σ) under the instrument operating conditions of these measurements was determined to be 6 pptV. In addition to the LOPAP, co-located measurements of NO, NO₂ and NOₓ (Thermo Scientific 42c - Mo convertor for NO₂ measurement, hence potentially responding to NOₓ interferences), and ozone (Thermo Scientific 49i) were performed. An optical particle spectrophotometer (TSI 3330) measured the particle number size distribution, and converted from particle number concentration (PNC) to total surface area (TSA) via the TSI AIM software (Version 9), which assumes particle sphericity. NOₓ, O₃ and particle data were obtained at a 1 min time resolution. Meteorological data (relative humidity, solar intensity, air temperature, and wind speed and direction) was obtained at 1 min time resolution from a nearby weather station operated by the University of Birmingham / UK Met Office, located within 100 m of the EROS site. At the University of Birmingham site, the eclipse (20th March 2015) occurred from 08:25 to 10:40 GMT (first / last contact) with the maximum coverage (89% of the sun’s face) occurring at 9:31.

3 Approach to exploring the anticipated response of HONO during the solar eclipse

HONO abundance is expected to respond to rapid changes in light conditions (i.e. with and without the shading condition of the solar eclipse), to an extent dependent upon the dominant HONO source and sink mechanisms, and their respective variation with solar radiation levels. To explore the variation in HONO (and other species) abundance anticipated across the eclipse timeframe (ca. 2 hours) a series of simple simulations were performed representing different scenarios (mechanisms) for potential atmospheric processes / emissions that may form and remove HONO. We do not attempt to reproduce the detailed gas- or heterogeneous chemistry using a formal model or chemical mechanism – as many of the species that would be required were not measured (notably, no VOC data were available, precluding modelling HOₓ sources or sinks). However, with some assumptions (below), over the limited duration of the eclipse event, the time evolution of HONO can be calculated from its initial abundance, the time variation of various possible source term(s), and its chemical removal.
which is dominated (≥ 95%, see below) by photolysis. This allows us to explore the response of potential HONO source(s), parameterised in terms of the measured variation of sunlight, NOx abundance and aerosol particle parameters, by comparison with the observed HONO variation over the course of the eclipse.

The attenuation of photolysis frequencies was approximated using values obtained from the TUV model\(^1\) for clear-sky conditions, with the reduction in \(j\) values modelled as a 1-\(\pi\) sine wave. We note that this is an approximation, both to the actual photolysis frequencies and the geometric coverage of the solar disk, but one which is acceptable in the context of the analysis which follows. The calculated \(j_{\text{NO}_2}\) values prior to and subsequent to the eclipse were consistent with those that would be estimated using photo-stationary steady state (NOx/O\(_3\) abundance, neglecting HO\(_2\)/RO\(_2\) reactions) to within 5.6±1.5 % (pre-eclipse), 21±1.9 % (during the eclipse) and 6.4±2.0 % (post-eclipse). The different scenarios examined are described in Table 1. In each case, the calculation was initiated at 07:30 GMT, prior to the start of the eclipse, and the subsequent evolution of HONO calculated according to the various alternative mechanisms presented below. In general, each case may be represented via Equation (1), where \(S_x\) is the source term for HONO formation in each scenario:

\[
d[HONO]/dt = S_x - j_{\text{HONO}} [\text{HONO}] \quad (E1)
\]

The different HONO sources considered, \(S_x\), are listed in Table 1. In each case HONO concentrations were calculated by numerically evaluating the differential equations for HONO formation / removal, using time-dependent observed values of the input parameters (\(j_{\text{NO}_2}, [\text{NO}_2]\) etc), on a 60-second timestep.

A number of assumptions are inherent in this approach: Equation 1 neglects HONO sinks other than photolysis (i.e., reaction with OH). Using OH measurements performed near to the measurement site during a previous campaign\(^2,3\), the rate of loss of HONO due to reaction with OH is calculated to be 1.1 – 5.0 % (using [OH] of 2 × 10\(^6\) – 9 × 10\(^6\) molec cm\(^{-3}\)) of that due to photolysis, at the lowest photolysis frequency corresponding to the eclipse maximum, hence this approximation is reasonable. Equation 1 also neglects homogeneous sources of HONO – from the (slow) termolecular OH + NO chemistry, which we estimate would contribute <7 % of the HONO production necessary to maintain the observed levels (section 4.2 below) and would, to an extent, offset the HONO + OH reaction. We also discount the proposed, but uncertain, H\(_2\)O + NO\(_2^*\) mechanism\(^4\). A number of assumptions regarding advection / mixing are made: We assume air mass homogeneity over the timescale of the...
eclipse period, and neglect vertical mixing. The former assumption is likely justified considering the background nature of the measurement site and limited timespan of the analysis; recent measurements of vertical profiles of HONO in urban areas point to a broadly uniform profile near to the ground (< 200m), with some increase near ground level attributed to traffic source mixing and ground surface production – although the measurement period spans the early morning time when boundary layer breakup / enhanced vertical mixing may occur. Finally, each potential HONO source mechanism is considered in isolation, while in reality multiple mechanisms likely occur in parallel. These and other assumptions are considered further in the Discussion, below.

In each case, the initial HONO concentration \([HONO]_{t=0}\), and the source mechanism scaling constant \(C\) (Table 1) was determined from the measured values of HONO, NOx, \(j_{NO2}\) and total aerosol surface area (TSA) as appropriate. Both \([HONO]_{t=0}\) and \(C\) were optimised independently, for each source scenario, by minimising the sum of squares of residuals between calculated and observed HONO concentrations for the two 1-hour periods prior to and immediately following the eclipse period (from 07:30-08:30 and 10:40-11:40 respectively). We adopt the pre/post eclipse period to allow prediction of hypothetical, no-eclipse HONO concentrations; we test the sensitivity of this time period selection by repeating the fitting including both the pre/post and eclipse periods (Supplementary Information, Figure S1); while the optimised parameters vary slightly the conclusions regarding which mechanism(s) are consistent/inconsistent with the data are unchanged. All other species were set to their actual observed levels for the relevant point in time. It is important to note that we do not assume HONO is in steady state / equilibrium with its production and removal terms. For each source scenario, we repeat the analysis under two conditions – firstly for the actual (eclipse influenced) reduction in solar intensity (photolysis frequencies), and secondly, for comparison, a hypothetical scenario where no eclipse occurred and clear-sky photolysis (as derived from the changing SZA) applied throughout the 2 hour time period. The resulting predicted and observed HONO concentrations are shown in Figure 2, and discussed below.

4 Results & Discussion

4.1 Observations

During the eclipse, weather conditions were calm (wind speed of 0.5 ± 0.4 m/s) with a mean relative humidity of 76 ± 6 % and temperature of 6.7 ± 1.7 °C, which are typical for the time
of year in the UK. In particular, the usual rise in temperature with daytime was observed, which may have begun the process of boundary layer break-up, although there was little evidence for change in NO\textsubscript{x} or PM levels associated with this. Relatively little change in HONO concentration was observed during the eclipse period (Figure 1a), suggesting that closely balanced photochemical source and photochemical sink mechanisms may be dominating HONO abundance. Ozone and NO\textsubscript{x} were observed to be tightly coupled during eclipse period (08:25 am to 10:40 am GMT, Figure 1b), and to follow their expected photochemistry\textsuperscript{48}, with NO and O\textsubscript{3} levels falling and NO\textsubscript{2} rising. During the eclipse, little variation was observed in the particle TSA (Figure 1c), which was found to be reasonably well correlated with the HONO concentration (R\textsuperscript{2}=0.63), possibly suggesting a common source or dilution process. Vehicle emissions have been identified as a significant source of HONO in urban environments, where a low HONO/NO\textsubscript{x} ratio is commonly used as a proxy parameter to evaluate HONO emissions from traffic.\textsuperscript{3, 49, 50} In the present study, high HONO/NO\textsubscript{2} (1.9 – 4.1 \%) and HONO/NO\textsubscript{x} (1.0 – 1.8 \%) ratios were observed before, during and after the eclipse. Previous empirically determined emission ratios of HONO/NO\textsubscript{x} based on tunnel and car exhaust studies are between 0.3\% and 1.0 \% \textsuperscript{9, 51}, with recent evaluations at the high end of this range, e.g. 1.24 ± 0.35\% \textsuperscript{50} and 0.72 – 1.01\% \textsuperscript{33} (determined in Birmingham). Thus the measured HONO/NO\textsubscript{x} ratio is at the top end of / in excess of the ratio of expected for vehicle emissions – suggesting that these are not the sole source of HONO at this site.

4.2 Evaluation of HONO source scenarios

Figure 2 compares the observed (red line/points) and simulated (actual, eclipse condition – solid blue line; hypothetical no eclipse condition – dashed blue line) time evolution of the HONO abundance for each scenario considered. Case 1 considers a scenario where no HONO source exists; this clearly diverges from the observed behaviour, indicating that additional HONO sources are in fact present. Note that the fit (minimise RMS residual) is affected by inclusion of the second fitting period, after the eclipse – but in which there is little sensitivity to the initial HONO level, due to the rapid photolytic decay. The reaction of OH radicals with NO forms HONO (via R2); no OH data were available during these experiments, but OH levels may be estimated from prior observations at this location (Heard, et al.\textsuperscript{42} – taking the mean OH level for 09:30 from these measurements (\textasciitilde 5\times10^5\text{ molec cm}^{-3}), a steady state HONO level of 0.04 ppb is obtained, approximately a factor of 15 lower than that observed mid-eclipse, indicating that OH + NO is not responsible for HONO.
formation in isolation, and rather forms a small contributor to the total HONO source. The role of OH + NO as a secondary HONO production term, alongside other, larger mechanisms, is considered further below.

Previous work in urban areas has suggested a daytime source of HONO that scaled with NO\textsubscript{2} levels and sunlight, specifically \(j_{\text{NO2}}\). Case 2 considers a HONO source term which is directly proportional to \(j_{\text{NO2}}\). The predicted HONO behaviour broadly matches that observed, indicating that in reality, the predominant source is likely to scale with \(j_{\text{NO2}}\) or some similar measure of solar insolation. Dark conversion of NO\textsubscript{2} to HONO upon moist surfaces is a long-established HONO formation mechanism (e.g. Finlayson-Pitts, et al.), and Case 3 considers a HONO source term which is directly proportional to NO\textsubscript{2} – however this scenario shows a poor fit of predicted to observed HONO levels, indicating that the predominant source does not scale with [NO\textsubscript{2}] in isolation. Photosensitized conversion of NO\textsubscript{2} on surfaces containing organics has been identified as a significant daytime HONO source in a number of studies. To explore the effect of such reactions, a HONO source which scaled with the product of \(j_{\text{NO2}}\) and the NO\textsubscript{2} mixing ratio was explored in Case 4 (Figure 2d), where a significantly better fit was observed, indicating that a source term which scaled with \(j_{\text{NO2}}\times [\text{NO}_2]\) could account for the majority of HONO production in this environment.

Direct emission of HONO from vehicles is another potential source in urban areas. In an urban area, particle total surface area (TSA) can be used as a reasonably conserved tracer for vehicle emissions. Case 5 (Figure 2e) considers a scenario where the HONO source scales with the TSA; however this also does not match the measured HONO levels during the eclipse, implying that a source scaled to particle TSA alone - or direct vehicular emission of HONO in isolation - cannot explain the observed HONO concentrations in this environment. Similar results were observed when particle number concentration was used as a metric of traffic particulate matter emissions, as opposed to TSA (not shown). However, HONO production proportional to the product of \(j_{\text{NO2}}\) and TSA was a better match to observed (Case 6, Figure 2f).

Cases 7 and 8 (Figures 2g and 2h respectively) represent HONO formation via NO\textsubscript{2} conversion on the surface of aerosol particles, without (Case 7 / Figure 2g) and with (Case 8 / Figure 2h) photo-enhancement respectively. These represent secondary chemical (photochemical) sources involving aerosol particles. In these cases, the HONO production rate is assumed to scale with TSA x [NO\textsubscript{2}] and \(j_{\text{NO2}}\times \text{TSA} \times [\text{NO}_2]\), respectively. Case 7 did
not reproduce the observed behaviour, indicating that such heterogeneous aerosol conversion
of NO₂ to HONO, without photoenhancement, could not, in isolation, account for HONO
production in this environment. Case 8 showed significantly closer reproduction of the
measured HONO evolution under the eclipse conditions, giving the best fit (lowest RMS
residual) of all the scenarios considered, and indicating that a mechanism dependent upon
solar insolation (j_{NO₂}), NO₂ abundance and aerosol surface area is able to reproduce the
observed HONO behaviour across the eclipse. The similarity of eclipse- and non-eclipse
simulations reflects the balanced photolytic dependence – as observed for the actual HONO.
While we obtain the optimal fit for case 8, in comparison to (e.g.) case 2, the geometric
surface area of ground surfaces significantly exceeds that of aerosol particles (by at least a
factor of 16-fold, assuming an aerosol surface area of 300 μm² cm⁻³ and boundary layer
height of 200m, which may be representative of the measurement location early in the
morning – low winds, under an eclipse condition), and hence it may be likely that ground
surface-mediated photoenhanced conversion of NO₂ dominated HONO formation at this
location, potentially augmented by contributions from aerosol surfaces.

In reality, it is likely that a combination of mechanisms occur in parallel, and addition of the
OH + NO reaction, with OH levels assumed equal to those measured^{42} and scaled to the
relative solar intensity expected during the eclipse (which may underestimate OH, given the
importance of non-photolytic sources such as alkene ozonolysis at this location in wintertime
found previously by Heard, et al.^{42} resulted in a slight improvement to the statistical fit –
although increasing degrees of freedom would be expected to improve agreement. As a
further evaluation of the performance of scenario 8, we applied this analysis approach to data
from the days preceding and following the eclipse event (i.e. 19th and 21st March
respectively); the simulation was able to satisfactorily reproduce the observed HONO levels
(Supplementary Info, Figure S2). We note that in Cases 3, 4, 7 and 8 the NOₓ would vary due
to changing PSS in the (hypothetical) non-eclipse case. This approximation has the effect of
biasing the non-eclipse (hypothetical) simulations high (as the real NO₂ would be lower, in
the absence of the eclipse-derived photolysis attenuation).

4.3 Insight from temporal variation of HONO production
A further analysis was undertaken to derive the cumulative HONO production with time, i.e., to remove the effect of the loss of HONO through photolysis under both the real (eclipse) and hypothetical (non-eclipse) conditions. This quantity – hypothetically conserved HONO, \([\text{HONO}]_{\text{Conserved}}\) was calculated using Equation 2:

\[
[\text{HONO}]_{\text{Conserved}} = C_n + \sum_{t = n - (n - 1)}^n (j \times \text{HONO} \times \frac{(C_n + C_{n-1}) \times \Delta t}{2})
\]

(2)

Where the \([\text{HONO}]_{\text{Conserved}}\) estimates the total HONO concentration that has been formed / emitted up to each point in time, derived from the observations after accounting for photolysis. HONO removal through reaction with OH is neglected, as above. \(C_n\) and \(C_{n-1}\) represent the observed HONO at time \(t_n\) and \(t_{n-1}\), while \(\Delta t\) is the time difference. A modest difference in HONO\(_{\text{conserved}}\) between the eclipse and without-eclipse conditions can be seen (Figure 3a). Concentrating on the actual (eclipse) condition, the conserved HONO shows an approximately monotonic rise, indicating an approximately constant emission rate. This is at variance with traffic-related emissions, which would be expected to show the characteristic “rush hour” behaviour, i.e. peaking at 8 – 9 am. Figure 3b shows the HONO\(_{\text{conserved}}/\text{NO}_x\) ratio – an approximately constant ratio would be expected if direct emissions dominated production (of both species), assuming NO\(_x\) is conserved, however in fact significant variation with time is observed (the chemical NO\(_x\) lifetime is estimated to be 17.5 hours with respect to OH + NO\(_2\), using the measured OH from Heard, et al. 42, indicating that NO\(_x\) can be considered to be approximately conserved on our timescale of 3.5 hours). The lack of a traffic / rush-hour pattern in the HONO\(_{\text{conserved}}\) and HONO\(_{\text{conserved}}/\text{NO}_x\) ratio further supports the inference that direct vehicular/combustion emissions, in isolation, are not the dominant source of HONO at this location, with some additional, photochemical / heterogeneous atmospheric chemical term is required to account for the observed behaviour. However, the results from Scenarios 1-8, which point to involvement of NO\(_2\) in all HONO production mechanisms with a degree of involvement with the observed behaviour, reinforce the ultimate importance of vehicle emissions (as the overall source of NO\(_x\)).

5 Concluding Remarks

HONO, NO\(_x\), O\(_3\) and particle number size distributions were measured at an urban background location over the course of a near-total solar eclipse in Birmingham, UK. The observed NO, NO\(_2\) and O\(_3\) responded as anticipated from the well-understood atmospheric
photochemistry. HONO levels showed limited variation across the eclipse period; given HONO removal is overwhelmingly dominated by photolysis, this observation indicates that photolytic sources predominate in producing HONO at this location. A simplified approach was adopted to explore which factors were required within the HONO source mechanism, assuming direct and multiplicative proportionality, by comparing the observed and predicted HONO levels across the eclipse perturbation. Source terms lacking a photolytic component – proportional to NO₂ concentration, aerosol particle number or aerosol particle surface area – were not consistent with the observed behaviour. Source terms with an insolation dependence, here represented as $j_{NO_2}$, better replicated the measurements, with strong evidence for the involvement of NO₂, and the best agreement (in isolation) obtained for a combination of $j_{NO_2}$, NO₂ concentration and aerosol surface area. Both the observed source behaviour, and the temporal variation in the HONO$_{conserved}$ levels and the HONO$_{conserved}$/NOₓ ratio, were inconsistent with primary traffic emissions dominating HONO production in this urban background location – although of course road traffic is ultimately the dominant emission source for NOₓ, and hence the key HONO precursor NO₂ identified here. Accordingly, future vehicle fleet evolution, which is expected to lead to significantly lower urban NOₓ levels through improved emissions technologies and penetration of electric vehicles, will also lead to lower HONO concentrations, and hence lower local OH radical production and local atmospheric oxidation in urban boundary layer locations such as that considered here.

**Author Information**

**Ajit Singh** - School of Geography, Earth and Environmental Sciences, University of Birmingham, Birmingham, B15 2TT, UK. ORCID: orcid.org/0000-0003-0986-2064, Email: a.singh.2@bham.ac.uk

**Leigh R. Crilley** - Department of Chemistry, York University, Toronto, Canada. ORCID: orcid.org/0000-0003-2268-9956, Email: lcrilley@yorku.ca

**William J. Bloss** - School of Geography, Earth and Environmental Sciences, University of Birmingham, Birmingham, B15 2TT, UK. ORCID: orcid.org/0000-0002-3017-4461, Email: W.J.Bloss@bham.ac.uk

**Francis D. Pope** - School of Geography, Earth and Environmental Sciences, University of Birmingham, Birmingham, B15 2TT, UK. ORCID: orcid.org/0000-0001-6583-8347, Email: F.Pope@bham.ac.uk
Conflicts of interest
There are no conflicts to declare

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Figures and Tables
Figure 1 Time series of (a) measured HONO and solar intensity, (b) NO\textsubscript{x} and Ozone, and (c) Particle total surface area (TSA) and HONO/NO\textsubscript{2} ratio.
Figure 2 Comparison between measured (red points) and calculated HONO mixing ratio during the solar eclipse, using actual eclipse photolysis frequencies (solid blue line) and (hypothetical) non-eclipse photolysis frequencies (dotted blue line). Shading indicates eclipse duration from first to last contact. Source scenarios as defined in Table 1.
Figure 3  

a) Comparison between observed HONO mixing ratios, and HONO$_{\text{conserved}}$ (the HONO abundance calculated after accounting for photolytic loss) under eclipse and (hypothetical) non-eclipse scenarios,  

b) Variation in HONO$_{\text{conserved}}$/NO$_x$ ratio under actual (eclipse) and hypothetical (non-eclipse) conditions.
Table 1: Summary of HONO production scenarios and optimised parameter values (HONO source strength determined in ppb s\(^{-1}\) and NO\(_2\) mixing ratio in ppb).

| Cases        | Potential sources                       | HONO Source strength (S) / ppb s\(^{-1}\) | Scale factor (C) | Initial HONO ([HONO]\(_{t=0}\) / ppb) | Fit RMS residual / ppb |
|--------------|------------------------------------------|------------------------------------------|------------------|----------------------------------------|------------------------|
| Case 1       | No source term                           | \(S_1 = 0\)                              | N/A              | 1.2                                    | 0.325                  |
| Case 2       | Photolysis-related                       | \(S_2 = C \times j_{NO2}\)               | 7.2 \times 10\(^{-2}\) ppb | 1.04                                   | 0.118                  |
| Case 3       | NO\(_2\)-related                         | \(S_3 = C \times [NO_2]\)                | 1.57 \times 10\(^{-5}\) s\(^{-1}\) | 0.71                                   | 0.067                  |
| Case 4       | Source(s) related to photolysis and NO\(_2\) | \(S_4 = C \times j_{NO2} \times [NO_2]\) | 3.99 \times 10\(^{-3}\) | 1.0                                    | 0.096                  |
| Case 5       | Source(s) related to aerosol surface area (TSA) | \(S_5 = C \times TSA\)                   | 1.65 \times 10\(^{-6}\) ppb s\(^{-1}\) (\(\mu\text{m}^2\text{cm}^{-3}\))\(^{-1}\) | 0.68                                   | 0.078                  |
| Case 6       | Source(s) related to photolysis and TSA  | \(S_6 = C \times j_{NO2} \times TSA\)    | 4.42 \times 10\(^{-4}\) ppb (\(\mu\text{m}^2\text{cm}^{-3}\))\(^{-1}\) | 0.97                                   | 0.090                  |
| Case 7       | Source(s) related to NO\(_2\) and TSA   | \(S_7 = C \times [NO_2] \times TSA\)     | 7.61 \times 10\(^{-8}\) s\(^{-1}\) (\(\mu\text{m}^2\text{cm}^{-3}\))\(^{-1}\) | 0.67                                   | 0.105                  |
| Case 8       | Source(s) related to photolysis, NO\(_2\) and TSA | \(S_8 = C \times j_{NO2} \times [NO_2] \times TSA\) | 2.31 \times 10\(^{-5}\) (\(\mu\text{m}^2\text{cm}^{-3}\))\(^{-1}\) | 0.94                                   | 0.075                  |
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