Variation in Growth, Physiology, Yield, and Quality of Wheat under the Application of Different Zinc Coated Formulations

Qudsia Nazir 1,2, Xiukang Wang 3,*, Azhar Hussain 4,*, Allah Ditta 5,6 ∗, Ayesha Aimen 4, Ifra Saleem 1,2, Muhammad Naveed 2, Tariq Aziz 2, Adnan Mustafa 7 ∗ and Nalun Panpluem 8

Abstract: Zinc (Zn) is critical for the activity of many enzymes including involved photosynthetic CO2 fixation and indirectly involved in the production of growth hormones and internode elongation in crop plants. In this regard, a field experiment was conducted to investigate the comparative effectiveness of the Zn blended, Zn coated and bio-activated Zn coated urea on the growth, physiological, yield, and quality of wheat. Three types of urea were prepared including bio-activated Zn coated, Zn coated and Zn blended urea. The bio-activated Zn coated urea was prepared by inoculating the powdered organic material with Zn solubilizing bacterium (Bacillus sp. AZ6) and then this material was mixed with ZnO. This bioactive Zn was coated on urea at the rate to formulate 1.5% bio-activated Zn coated urea. Moreover, Zn blended urea was prepared by mixing powder ZnO with urea while Zn coated urea with 1.5% Zn was prepared by mixing ZnO and urea under proper moisture conditions to ensure proper coating. In results, growth parameters were significantly increased with the application of bio-activated Zn coated urea as compared to other urea formulations and the control. The same treatment caused the maximum increase in quality parameters like oil contents (55%), protein (30%), and N concentration (30%) as compared to the control. In conclusion, the application of 1.5% bio-activated Zn coated urea was highly effective in enhancing the growth, physiological, yield, and quality parameters of wheat.

Keywords: zinc; Zn biofortification; zinc solubilizing bacteria; bioactivated zinc; urea

1. Introduction

Zn plays a critical role in various physiological processes (maintains the structural integrity of rubisco, influx rate of K+ in cells, and opening of stomata) of crop plants [1–3]. In plants, Zn affects water uptake and its transport [4] and reduces the adverse effects of heat [5] and salinity stress [6] by inducing the production of H2O2-scavenging glutathione reductase and ascorbate peroxidase [7]. Zn is required for the synthesis of tryptophan, which is a precursor of indole acetic acid, protein, and nucleic acid synthesis, energy production, carbohydrate and lipid metabolisms [1,3].
Wheat is the most vital cereal and staple crop cultivated in Pakistan [8]. It contributes about 1.7% to gross domestic product (GDP). During 2019–2020, wheat was grown on an area of 8825 '000 ha with 24,946 '000 tones production [9]. In the cereal-based cropping system, Zn deficiency is a widespread chronic problem among the human population and is expected to increase up to 138 million by 2050. More than 0.1 million deaths, especially in young children, are expected due to diarrhea and pneumonia caused by Zn deficiency [10].

Being an essential micronutrient, Zn plays an important role in improving the growth, physiological, yield, and quality parameters of various crops. However, Zn deficient soils are widespread in subtropical regions of the world and the main reason behind this deficiency is the cereal-based cropping system prevailing in these regions. Moreover, plants grown on such soils show Zn deficiency, and overall plant growth, physiology, and quality are disturbed. Pakistani soils are Zn deficient due to more CaCO$_3$ contents with high soil pH and less organic matter [11–13]. Due to Zn deficiency, the crops, especially cereal crops, cultivated on such soils have low Zn contents in grains. Therefore, it is a need of the hour to find out some sustainable and cost-effective approaches to combat Zn deficiency in soils and ultimately in food grains.

Among various strategies, biofortification through the soil and/or foliar application of Zn through chemical fertilizers, seed coating, impregnated fertilizers, etc., are mainly practiced [14]. Zinc application in soil not only increases the growth and yield of plants but also improves overall vigor and quality parameters such as sugars and oil contents [15]. However, due to the high cost involved and environmental degradation caused, it is not an eco-friendly strategy. Moreover, coating of natural materials on different fertilizers has also been the interest of scientists in recent studies aimed at improving the fertilizer use efficiency and subsequent improvement of crop yields [16,17].

Fertilizer management is a technique aimed to improve plant vigor, quality, and plant physiology under various cropping systems [18,19]. The most commonly used source of Zn in Pakistan is ZnSO$_4$, but due to its high cost and scarcity or even unavailability in the market, farmers are reluctant to use it. ZnO is an effective source as it contains 80% Zn contents but in an insoluble form. This insoluble Zn can be solubilizing by ZSB (zinc solubilizing bacteria). Zn solubilizing bacteria are capable of enhancing Zn bioavailability in the soil through the secretion of organic acids, chelating agents, and siderophores [20–24]. Moreover, other plant growth promotion activities possessed include the solubilization of nutrients such as phosphorus, alleviation of ethylene stress through ACC-deaminase activity, and production of phytohormones to enhance root growth for efficient uptake of nutrients, acting as biocontrol agents, etc. [25–27]. By keeping the above-mentioned facts in mind, a field experiment was conducted in which the impact of different Zn formulations, i.e., ZnO coated, blended and bio-activated Zn coated urea, were investigated on the growth, yield, quality, and physiological parameters of wheat.

2. Materials and Methods

A field experiment was conducted in the farm area of the Institute of Soil and Environmental Sciences, University of Agriculture, Faisalabad, Pakistan using wheat as a test crop. The comparative effectiveness of Zn blended, Zn coated, and bio-activated Zn coated urea was evaluated for physiological, quality, and biochemical parameters under field conditions for the wheat crop (FSD-2008). The weather data regarding average temperature (°C) and rainfall (mm) during the cropping season from October 2014 to April 2015 is presented in Figure 1.

2.1. Soil Analysis

Composite soil samples were taken randomly from the experimental field, air-dried, ground, sieved (2 mm), and analyzed regarding various physicochemical characteristics following standard procedures [28–32]. The soil sandy clay loam in texture (sand 52.3%; silt 28.5%; clay 19.2%) with pH = 7.85, electrical conductivity (EC) = 1.43 dS m$^{-1}$, organic
matter = 0.69%, total nitrogen = 0.07%, available phosphorus = 8.6 mg kg\(^{-1}\); extractable potassium = 85 mg kg\(^{-1}\); and plant available Zn = 0.66 mg kg\(^{-1}\).

![Graph: Average temperature (°C) and rainfall (mm) during the cropping season from October 2014 to April 2015.](image)

**2.2. Preparation of Different Urea Formulations**

Earlier, we isolated zinc-solubilizing bacteria from the maize rhizosphere by using the dilution plate technique on nutrient agar medium [33]. The inoculum of the strain AZ6 was prepared by growing it in a 1000 mL conical flask containing Bunt and Rivera basal medium [34]. The inoculated flasks were incubated (28 ± 1 °C) for 72 h in the orbital shaking incubator at 100 rpm. Before using inoculum, an optical density of 0.5 at 535 nm was maintained with \(10^8\)–\(10^9\) colony-forming units (CFU) mL\(^{-1}\). This suspension (inoculum) of the bacterial strain was used for urea coating to prepare bio-activated Zn coated urea [20]. Briefly, powdered organic material (plant residues) was first dried in an oven at 80 °C, inoculated with bacterial strain AZ6, and incubated for 72 h at 30 ± 1 °C in an incubator. Then, this bio-augmented organic material was thoroughly mixed with 300–400 mesh size ZnO in the ratio of 40:60 \(\text{w/w}\) (powder ZnO: bio-augmented organic material). This mixture was again incubated for 3 days at 30 ± 1 °C to achieve maximum chelation of Zn with organic complexes. The bio-activated Zn was coated on urea at 1.5\% \(\text{w/w}\) to prepare bio-activated Zn coated urea. Before impregnation/coating on urea granules, the bio-activated Zn complex was once again passed through 300-400 mesh size sieves. All the precautions were used and there was no change in the composition of urea. To prepare Zn coated urea, bacterial strain AZ6 was not used and Zn \(1.5\% \text{w/w}\) was coated using ZnO with proper moisture contents while Zn blended urea was prepared by mixing a certain amount of ZnO at 1.5\% Zn \(\text{w/w}\) [35].

**2.3. Experimental Description**

The field experiment was conducted with six treatments including \(T_0 = \text{control (no Zn)}\), \(T_1 = \text{ZnSO}_4 (\text{recommended})\), \(T_2 = \text{ZSB (zinc solubilizing bacteria)}\), \(T_3 = 1.5\% \text{Zn coated urea}\), \(T_4 = 1.5\% \text{bio-activated Zn coated urea}\), \(T_5 = 1.5\% \text{Zn blended urea}\). The treatments were arranged following randomized complete block design (RCBD) in triplicate. Seeds of wheat variety FSD-2008 were purchased from Ayub Agriculture Research Institute, Faisalabad, and sown in the field using a hand drill. Each experimental unit had an area of 35 m\(^2\). Chemical fertilizers such as urea, di-ammonium phosphate, and sulfate of potash were used as NPK sources at 160:110:90 kg ha\(^{-1}\). Zinc was applied at the rate of 5 kg ha\(^{-1}\). Recommended agronomic practices such as weeding, hoeing, etc. were
performed. The crop was harvested at maturity and data regarding growth, yield, and biochemical parameters were taken.

2.4. Growth and Yield Parameters

The plant height and spike length were measured with the help of a measuring tape from top to bottom. No. of tillers plant$^{-1}$ were counted manually. At maturity, the yield parameters such as 100-grain weight, grain yield, and biomass production per hectare of wheat were recorded.

2.5. Physiological Parameters

2.5.1. Electrolyte Leakage

Electrolyte leakage was determined by the method described by [36]. Leaf samples were washed with double deionized water (DDW) to remove any surface contamination, cut into leaf discs, put into the closed vial containing DDW (10 mL), incubated on a rotatory shaker for 24 h, and electrical conductivity of the solution ($EC_1$) was determined by using an electrical conductivity meter (Jenway 4070, ELE, England, UK). Then, vials with leaf samples were autoclaved at 120 °C for 20 min., cooled, and electrical conductivity ($EC_2$) was again measured. The following formula was used to calculate electrolyte leakage:

\[
\text{Electrolyte leakage (\%)} = \frac{EC_1}{EC_2} \times 100
\]

2.5.2. Carbonic Anhydrase Activity

Carbonic anhydrase (CA) activity was determined by following the method proposed by [37]. Briefly, leaf samples were cut into small pieces (1 cm$^2$), mixed, and 200 mg leaf pieces were weighed and suspended in 0.2 M cysteine hydrochloride solution. The samples were incubated at 40 °C for 20 min., blotted, and transferred to the test tubes containing phosphate buffer (pH 6.8), followed by the addition of 0.2 M alkaline bicarbonate solution and 0.002% bromothymol blue indicator, and incubated at 50 °C for 20 min. After the addition of 0.2 mL of methyl red indicator, the reaction mixture was titrated against 0.05 N HCl. The CA activity was calculated and expressed as \( \mu \text{mol CO}_2 \ \text{kg}^{-1} \ \text{FW} \ \text{s}^{-1} \).

2.6. Quality Parameters

2.6.1. Oil Contents

The oil contents in each flour sample were determined using the Soxhlet apparatus [38]. Briefly, five grams of flour was extracted with petroleum ether and condensed at the rate of 2–3 drops s$^{-1}$ for 8 h. After distillation of excess ether, the residue remaining in the extraction flask was dried at 100 °C for half an hour until a constant weight was achieved. Oil contents were calculated by using the following formula:

\[
\text{Oil contents (\%)} = \frac{\text{Weight of ether extract}}{\text{Weight of flour sample}} \times 100
\]

2.6.2. Ash Contents (Mineral Contents)

The ash contents in each grain sample were determined by following the procedure mentioned in AACC [38]. The whole grain sample taken in pre-weighed crucibles were on Bunson burner before incinerating in the muffle furnace where a temperature of 550 °C was maintained until the sample converted to gray whitish residue.

\[
\text{Ash contents (\%)} = \frac{\text{Weight of residue}}{\text{Weight of sample}} \times 100
\]
2.6.3. Nitrogen and Crude Protein in Grain Samples

Nitrogen contents in wheat grains samples were determined by Kjeldahl method [39]. The crude protein concentration was calculated by multiplying nitrogen concentration in grain samples with a conversion factor of 5.7 [40].

2.6.4. Zinc Contents in Grain Samples

Air-dried and ground, grain sample (1 g) was placed in the digestion flask and tri-acid (HNO₃, H₂SO₄, HClO₄) reagent (10 mL) was added and held overnight. The samples were heated on the hot plate carefully the next day until the production of red NO₂ fumes was ceased. The flasks were allowed to cool down and then 2–4 mL HClO₄ (70%) was added. The samples were heated again to reduce volume via evaporation. When the vapors were condensed, contents of the flask were transferred to a volumetric flask (50 mL) and volume was made up to the mark with DDW. For control, the same procedure was performed except it was without a grain sample. Samples were then filtered and used for Zn determination via atomic absorption spectrophotometer (PerkinElmer, Analyst 100, Waltham, MA, USA). A calibration curve was prepared by using working standards prepared from the stock solution. The stock solution was made by dissolving ZnSO₄·7H₂O (4.398 g L⁻¹). The sub-stock solution of 100 ppm was prepared by taking 10 mL stock solution in a 100 mL flask. The volume was made with DDW. The working standards of 0.5, 1, 1.5, 2, 2.5, and 3 mg L⁻¹ were prepared by taking sub-stock solution.

2.6.5. Statistical Analysis

A computer-based software i.e., Statistix v. 8.1 (Analytical Software, Tallahassee, FL, USA) was used to calculate analysis of variance for the collected data and treatment means were compared by least significant difference (LSD) test at α = 0.05 [41].

3. Results

3.1. Growth Parameter

All treatments i.e., Zn coated, Zn blended, and bio-activated Zn coated urea showed a statistically significant (p < 0.05) effect on growth parameters but the maximum increase was recorded in the treatment where 1.5% bio-activated Zn coated urea was applied. Figure 2a shows a significant effect of Zn coated urea on plant height as compared to the control treatment where no Zn was applied. With the application of 1.5% bio-activated Zn coated urea (87.5 cm), an increase of 25.33 and 2.6% was recorded in plant height as compared to the control (65.3 cm) and recommended Zn as ZnSO₄ (85.16 cm) treatments, respectively. With the application of recommended Zn, 1.5% Zn coated (86.8 cm) and 1.5% Zn blended urea (80.7 cm) showed 29.23 and 33% increase, respectively. On the other hand, 27.51% increase in spike length was recorded with the application of recommended Zn (10.25 cm) as compared to the control. The application of 1.5% bio-activated Zn coated urea showed an increase of 0.8 and 8.5% in plant height as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

Bio-activation and coating of Zn on urea showed a significant effect (p < 0.05) on spike length of wheat as shown in Figure 2b. The data clearly showed that the application of 1.5% bio-activated Zn coated urea (11.25 cm) caused the maximum increase (33.95%) compared to the control (7.43 cm). The application of ZSB (8.50 cm) showed an almost 29% increase over the control. The application of 1.5% Zn coated (10.50 cm) and 1.5% Zn blended urea (9.50 cm) showed 29.23 and 33% increase, respectively. On the other hand, 27.51% increase in spike length was recorded with the application of recommended Zn (10.25 cm) as compared to the control. The application of 1.5% bio-activated Zn coated urea showed 8.8% increase in spike length as compared to recommended Zn as ZnSO₄. The application of 1.5% bio-activated Zn coated urea caused an increase of 7.1 and 18.4% in spike length as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.
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![Figure 2](image-url)

Figure 2. Growth parameters of wheat: (a) plant height; (b) spike length; (c) No. of tillers per plant, grown with different formulations of urea. Where $T_0$ = No Zn; $T_1$ = recommended Zn (ZnSO$_4$); $T_2$ = ZSB; $T_3$ = 1.5% Zn coated urea (ZnO); $T_4$ = 1.5% bio-activated Zn coated urea (ZnO); and $T_5$ = 1.5% Zn blended urea (ZnO). Bars showing the same letters do not differ significantly ($p < 0.05$).
The impact of Zn application on the no. of tillers plant$^{-1}$ is presented in Figure 2c. The data indicated that all treatments showed a statistically significant effect ($p < 0.05$) on the no. of tillers plant$^{-1}$ as compared to the control (no Zn). For example, alone application of ZSB (4) showed 25% increase in no. of tillers plant$^{-1}$ as compared to the control while the maximum increase (57.14%) was recorded with the application of 1.5% bio-activated Zn coated urea (7), followed by ZSB (42.85%), 1.5% Zn blended urea (28.57%), recommended Zn (14.28%) and 1.5% Zn coated urea (7.14%) compared to the control. The application of 1.5% bio-activated Zn coated urea caused an increase of 7.7 and 40.0% in no. of tillers plant$^{-1}$ as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

3.2. Physiological Parameters

Regarding the effect of Zn application in the form of Zn coated, Zn blended, bio-activated Zn coated urea, and ZSB (Bacillus sp.), a statistically significant ($p < 0.05$) effect on electrolyte leakage of wheat was observed in Figure 3a.

The maximum electrolyte leakage was observed in the ZSB (38.5%) and the control where no Zn was applied (38.4%) and that of the minimum value (20.2%) was recorded with the application of 1.5% bio-activated Zn coated urea, which was 40% less as compared to the control treatment. Similarly, 1.5% Zn blended urea (25.8%) caused a decrease of 8.7% in electrolyte leakage as compared to the control. The application of 1.5% bio-activated Zn coated urea caused a decrease of 17.7 and 21.8% in electrolyte leakage as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

In the case of carbonic anhydrase (CA) activity, the maximum (321.67 μmol CO$_2$ kg$^{-1}$ s$^{-1}$) was observed in the treatment where 1.5% bio-activated Zn coated urea was applied and it was 34% more as compared to the control (212.33 μmol CO$_2$ kg$^{-1}$ s$^{-1}$) Figure 3b. The application of ZSB resulted in 220 μmol CO$_2$ kg$^{-1}$ s$^{-1}$ CA activity, which was 4.3% more compared to the control. The treatments where 1.5% Zn coated and 1.5% Zn blended urea was applied showed 286.67 and 284.0 μmol CO$_2$ kg$^{-1}$ s$^{-1}$ CA activities, respectively. Recommended Zn as ZnSO$_4$ (273.33 μmol CO$_2$ kg$^{-1}$ s$^{-1}$) showed an almost 21% increase in CA activity compared to the control. The application of 1.5% bio-activated Zn coated urea caused an increase of 12.2 and 13.3% in CA activities as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

Figure 3. Comparative effectiveness of different Zn formulations of urea regarding electrolyte leakage and carbonic anhydrase activity of wheat: (a) = Electrolyte leakage (%) and (b) = carbonic anhydrase activity. Treatments are as, T$_0$ = No Zn; T$_1$ = recommended Zn (ZnSO$_4$); T$_2$ = ZSB; T$_3$ = 1.5% Zn coated urea (ZnO); T$_4$ = 1.5% bio-activated Zn coated urea (ZnO); and T$_5$ = 1.5% Zn blended urea (ZnO). Means sharing the same letters do not differ significantly ($p < 0.05$).
to the control treatment. Similarly, 1.5% Zn blended urea (25.8%) caused a decrease of 8.7% in electrolyte leakage as compared to the control. The application of 1.5% bio-activated Zn coated urea caused a decrease of 17.7 and 21.8% in electrolyte leakage as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

In the case of carbonic anhydrase (CA) activity, the maximum (321.67 μmol CO₂ kg⁻¹ s⁻¹) was observed in the treatment where 1.5% bio-activated Zn coated urea was applied and it was 34% more as compared to the control (212.33 μmol CO₂ kg⁻¹ s⁻¹) Figure 3b. The application of ZSB resulted in 220 μmol CO₂ kg⁻¹ s⁻¹ CA activity, which was 4.3% more as compared to the control. The treatments where 1.5% Zn coated and 1.5% Zn blended urea was applied showed 286.67 and 284.0 μmol CO₂ kg⁻¹ s⁻¹ CA activities, respectively. Recommended Zn as ZnSO₄ (273.33 μmol CO₂ kg⁻¹ s⁻¹) showed an almost 21% increase in CA activity compared to the control. The application of 1.5% bio-activated Zn coated urea caused an increase of 12.2 and 13.3% in CA activities as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

3.3. Yield Parameters

A statistically significant effect (p < 0.05) of different Zn formulations on yield parameters of wheat under field conditions compared to the control was observed (Figure 3). The plots treated with 1.5% bio-activated Zn coated urea showed 15.75 tons ha⁻¹ biomass, which was 4% more as compared to the control (13.47 tons ha⁻¹ biomass), Figure 4a. The treatments with 1.5% Zn coated and 1.5% Zn blended urea showed 14.83 and 15.13 tons ha⁻¹ biomass, respectively and both were about 2% more as compared to the control. In comparison to the recommended ZnSO₄ (14.5 tons ha⁻¹ biomass), the application of 1.5% bio-activated Zn coated urea caused about 0.6% increase in biomass. A similar trend was recorded regarding the grain yield of wheat, Figure 4b. The application of 1.5% bio-activated Zn coated urea caused an increase of 6.2 and 4.1% in biomass as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

The data regarding grain yield (tons ha⁻¹) of wheat showed that a statistically significant increase with the application of different Zn formulations, Figure 4b. The application of 1.5% bio-activated Zn coated urea yielded 5.60 tons ha⁻¹ grains and this was 33% increase as compared to the control (3.42 tons ha⁻¹ grains). Almost similar results were obtained with the application of 1.5% Zn coated and 1.5% bio-activated Zn coated urea applied. With the application of 1.5% Zn blended urea (4.67 tons ha⁻¹ grains), about 28% increase in grain yield was observed as compared to the control. About 3.5% increase was observed with the application of recommended Zn as ZnSO₄ (3.92 tons ha⁻¹ grains) compared to the control treatment. The application of 1.5% bio-activated Zn coated urea caused an increase of 9.8 and 19.6% in grain yield as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

Figure 4. Cont.
Figure 4. Comparative effectiveness of different Zn formulations of urea regarding yield parameters of wheat (a) Biomass (tons/ha), (b) Grain yield (tons/ha), and (c) 100 grains weight. Where $T_0 = \text{no Zn}; T_1 = \text{recommended Zn (ZnSO}_4\text{)}; T_2 = \text{ZSB}; T_3 = \text{1.5% Zn coated urea (ZnO)}; T_4 = \text{1.5% bio-activated Zn coated urea (ZnO)}; \text{and T}_5 = \text{1.5% Zn blended urea (ZnO)}$. Means sharing the same letters do not differ significantly ($p < 0.05$).

Data presented in Figure 4c showed that different Zn formulations have a statistically significant effect ($p < 0.05$) on the 100-grain weight of wheat. With the application of recommended Zn as ZnSO$_4$ (42.19 g), 19% increase in 100-grain weight was observed compared to the control (34.79 g). Application of 1.5% bio-activated Zn coated urea resulted in the maximum (44.87 g) 100-grain weight compared to all other treatments. The treatments with 1.5% Zn coated (42.47 g) and 1.5% Zn blended urea (39.25 g) showed 18.6% and 11% increase in 100-grain weight as compared to the control (no Zn). The application of 1.5% bio-activated Zn coated urea showed 6% and 12.5% increase as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively. The alone use of ZSB (38.81 g) caused 11% increase in 100-grain weight with respect to the control. The application of 1.5% bio-activated Zn coated urea caused an increase of 5.6% and 14.3% in 100-grain weight as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

3.4. Quality Parameters

Data regarding the impact of different Zn formulations after the bio-activation and coating of ZnO on different quality parameters of wheat is presented in Table 1. Regarding ash (minerals) contents (%), the coating and bio-activation of Zn at the level of 1.5% resulted in the maximum increase compared to the control with no Zn application. The application of 1.5% bio-activated Zn coated urea and recommended Zn (ZnSO$_4$) showed almost similar results regarding ash contents i.e., 6.06 and 5.70%, respectively. The treatments where
1.5% Zn coated and 1.5% Zn blended urea were applied showed 5.0 and 4.1% ash contents. ZSB (Bacillus sp.) and the control showed 4 and 3% ash contents. In all these treatments, 1.5% bio-activated Zn coated urea showed 52% increase over the control while 19 and 30% increase as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively. The application of 1.5% bio-activated Zn coated urea caused an increase of 23.7 and 43.8% in ash contents as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

Table 1. Comparative effectiveness of the optimum level of zinc blended, zinc coated, and bioactivated zinc-coated urea to improve quality parameters under field conditions in wheat.

| Treatment                      | Ash Contents (%) | Oil Contents (%) | N (%)   | Protein (%) | Zn (µg g⁻¹ DW) |
|--------------------------------|------------------|------------------|---------|-------------|----------------|
| No Zn                          | 2.9 ± 0.09 c     | 0.04 ± 0.004 b   | 1.80 ± 0.02 c | 10.5 ± 0.23 d | 34.5 ± 0.2 f   |
| ZnSO₄                          | 5.7 ± 0.09 ab    | 0.07 ± 0.004 ab  | 2.50 ± 0.09 a | 14.3 ± 0.2 b  | 43.5 ± 0.3 b   |
| ZSB                            | 4.0 ± 0.09 bc    | 0.07 ± 0.002 ab  | 2.20 ± 0.04 b | 12.5 ± 0.3 c  | 37.5 ± 0.2 e   |
| 1.5% Zn coated urea            | 5.0 ± 0.3 ab     | 0.07 ± 0.002 ab  | 2.40 ± 0.04 a | 13.9 ± 0.3 b  | 38.2 ± 0.09 d  |
| 1.5% bio-activated Zn coated urea | 6.0 ± 0.4 a   | 0.09 ± 0.002 a   | 2.60 ± 0.14 a | 14.9 ± 0.2 a  | 45.2 ± 0.09 a  |
| 1.5% Zn blended urea           | 4.1 ± 0.04 bc    | 0.05 ± 0.002 b   | 2.43 ± 0.03 ab | 13.8 ± 0.2 b  | 42.2 ± 0.094 c |
| LSD                            | 1.8047           | 0.0318           | 0.2374   | 0.5976       | 0.3986         |

Note: Values show means ± S.E. where n = 3. Means sharing the same letters within the column do not differ significantly (p < 0.05).

In the case of oil contents, all treatments showed significant effects however, the maximum increase (55%) was observed with the application of 1.5% bio-activated Zn coated urea as compared to the control where no Zn was applied. Similarly, the application of 1.5% Zn coated and 1.5% Zn blended urea showed an increase of 39 and 21% in oil contents, respectively as compared to the control. In comparison to recommended Zn (ZnSO₄), an increase of 33% was recorded with the application of 1.5% bio-activated Zn coated urea. The application of 1.5% bio-activated Zn coated urea caused an increase of 23.4 and 66.7% in oil contents as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

A statistically significant effect of different Zn formulations was observed in the case of nitrogen (N) contents in wheat grains (Table 1). It was clear from the data that the application of 1.5% bio-activated Zn (ZnO) coated urea was more effective and it caused an increase of 30% in protein contents compared to the control. 1.5% Zn coated and 1.5% Zn blended urea showed an increase of 24.18 and 23.86%, respectively in N contents as compared to the control. The recommended Zn (ZnSO₄) showed 2.50% N contents that were 1.8% more as compared to the control. As compared to recommended Zn (ZnSO₄), 1.5% bio-activated Zn coated urea showed an increase of 21.8% in N contents as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

The effect of different Zn formulations on the grain Zn concentration of wheat grains is shown in Table 1. The data clearly showed that all treatments showed statistically significant effects (p < 0.05) as compared to the control. The use of 1.5% bio-activated Zn (ZnO) coated urea was more effective and it caused an increase of 30% in protein contents compared to the control. It was followed by recommended Zn (ZnSO₄) with 14.3%, 1.5% Zn coated with 13.9%, 1.5% Zn blended urea with 13.8%, ZSB (Bacillus sp.) with 12.5% and the control with 10.5% protein contents. The application of 1.5% bio-activated Zn coated urea caused an increase of 6.4 and 7.8% in protein contents as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

The effect of different Zn formulations on the grain Zn concentration of wheat grains is shown in Table 1. The data clearly showed that the application of Zn in different forms significantly (p < 0.05) increased the grain Zn concentration. The use of 1.5% bio-activated Zn coated urea showed 23.60% increase in grain Zn concentration as compared to the control. While the treatments with 1.5% Zn coated and 1.5% Zn blended urea resulted in 9.60 and 17.85% increase in grain Zn concentration as compared to the control. The recommended Zn application (ZnSO₄) increased grain Zn concentration up to 20.6% as
compared to the control. The bacterium ZSB i.e., Bacillus sp. AZ6 increased Zn contents up to 8% with respect to the control. As compared to recommended Zn (ZnSO₄), 3.6% increase in grain Zn concentration was observed with 1.5% bio-activated Zn coated urea. The application of 1.5% bio-activated Zn coated urea caused an increase of 18.3 and 7.1% in grain Zn concentration as compared to 1.5% Zn coated and 1.5% Zn blended urea, respectively.

4. Discussion

Zinc, being a micronutrient, is involved in a wide variety of physiological processes [2]. It is required for optimum growth, yield, and quality of crop produce. Different approaches have been adopted to supply Zn to the plants such as soil, foliar, and soil + foliar application of chemical fertilizers, organic fertilizers, etc. The main issues with chemical fertilizers are their detrimental effects on soil ecology and lower solubility in soil, which make these approaches not feasible for poor farmers. On other hand, the use of PGPR (plant growth-promoting rhizobacteria) is an economical and environment friendly approach. The use of bio-inoculants such as bacteria could be a viable option to enhance Zn bioavailability and to achieve the objective of low input and sustainable agriculture to overcome Zn deficiency in soils and plants [42].

In soils, about 90% of Zn is present in an insoluble form, which is not available for plant uptake. On the other hand, metal salts solubilization is an important feature of so-called PGPR. Normally, PGPR mobilizes the nutrients by the secretion of organic acids, chelating agents (siderophores), and through exchange reactions [43]. Various reports have confirmed that PGPR secretes siderophores, gluconate, or the derivatives of gluconic acids, e.g., 2 and 5 ketogluconic acid, and various other organic acids for the mobilization of Zn and iron in soil [44–47]. Use of Zn solubilizing PGPR to enhance the growth and yield of crop plants is expected to become an emerging trend in contemporary agriculture in the near future. In the same way, bio-activation of an insoluble source of Zn i.e., ZnO to enhance Zn bioavailability and then the coating of this bio-activated Zn (ZnO) on urea could be an effective and sustainable approach to enhance Zn bioavailability in soil and biofortification of Zn in wheat grains, thereby improving physiology and quality of crops. Prasad and his coworkers reported in 2013 that the major benefit of Zn coated urea is saving the amount of Zn to be applied, only 2.83 kg Zn ha⁻¹ was required with Zn coated urea as against 6 kg Zn ha⁻¹ in the case of soil + foliar application of ZnSO₄. The Zn coated urea is therefore a favorable fertilizer in developing countries with small landholding farmers [48].

ZnO is sparingly soluble and not readily available to crop plants but through bio-activation and coating of ZnO at the level of 1.5%, the maximum response was recorded regarding growth, yield, and physiological parameters of wheat under both field conditions in the present study. This problem could be overcome via bio-activated Zn coated urea as it controls the formation of insoluble Zn compounds due to the presence of organic acids. In the previous studies, the effective results were achieved by the application of 2% ZnO coated urea in the wheat-rice cropping system in the present investigation, the maximum increase in most of the growth, physiological, yield, and quality parameters of wheat recorded was observed with the application of 1.5% bio-activated (Zn) ZnO coated urea.

The physiological parameters such as carbonic anhydrase (CA) activity in which Zn act as a cofactor improved significantly with the application of Zn in Zn deficient soils, especially in rice-wheat cultivated areas [49]. Carbonic anhydrase activity decreases in many plants under Zn deficit conditions [50]. CA activity is directly related to the Zn concentration in plants. Under severe Zn deficient conditions, no activity of CA was observed [51]. For more activity of CA in the mesophyll cells, Zn application is necessary. The activity of CA is an indicator for the levels of physiologically active Zn [50]. Due to the involvement of CA in CO₂ fixation during photosynthesis, Zn application significantly improved photosynthesis, which automatically increased all other growth, physiological, and yield parameters [52]. The ZSB used in the present study could produce auxin, which might have increased the root growth of plants and ultimately increased the nutrients
uptake [53]. Zinc has a great role in membrane permeability; electrolyte leakage is the direct measurement of membrane permeability. With the application of Zn, the decreased value of electrolyte leakage was observed but the maximum reduction in electrolyte leakage was observed in the plots, which received 1.5% bio-activated Zn coated urea.

The cereals (wheat, rice, maize, etc.) are important food crops in developing countries, cultivated on large areas. Cereals being a staple food for the majority of people, the quality parameters like proteins, nitrogen, oil, ash contents in grains in wheat need to be assessed. In the present study, these parameters were significantly improved with the application of Zn and the maximum increase was observed with 1.5% bio-activated Zn coated urea. These results conform with the findings of [54,55]. The increase in quality parameters might be due to the contribution of Zn during photosynthesis, starch, and carbohydrate metabolism. Zn also starts glutamic dehydrogenase activity, RNA and DNA synthesis, which enhance gluten accumulation during the later stages of grain filling [55,56]. As N and Zn have a synergistic effect so the proper application of Zn improved N concentration in grains was obtained [57]. In the present study, dry matter (%) increased while moisture (%) decreased with Zn fertilization and these results are in agreement with the [58]. It has also been reported that oil contents are increased with the application of proper Zn [59].

The yield parameters such as grains yields and biomass production increased significantly with the application of Zn and the maximum increase was recorded with the application of 1.5% bio-activated Zn coated urea. Our results are in line with Zeb et al. [20] who reported improved yield parameters with the application of Zn. Grain production is an important parameter contributing towards yield, with the application of Zn, a significant increase in grain yield of wheat was observed in the present study.

5. Conclusions

Bio-activation and coating of Zn on urea to supply Zn to plants is a novel, economical, and eco-friendly approach. It also saves farmers time as compared to the application of Zn and urea separately. The application of 1.5% bio-activated Zn coated urea has a marked effect not only on plant growth, yield, and physiological parameters but also on Zn biofortification of wheat under field conditions. Moreover, cereals, especially wheat grown on Zn deficient soil, 1.5% bio-activated Zn (ZnO) coated urea is found effective in improving the Zn contents and quality parameters of wheat grains. Therefore, the application of bio-activated Zn (ZnO) coated urea could be a sustainable approach for Zn biofortification of wheat under farmers’ field conditions. However, to warrant its practical application, rigorous studies under variable field conditions are required to authenticate its efficacy in the future.

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