Abstract Root-knot nematodes (Meloidogyne spp.) are obligate endoparasites of a wide range of plant species. The infective stage is attracted strongly to and enters host roots at the zone of elongation, but the compounds responsible for this attraction have not been identified. We developed a simple assay to investigate nematode response to chemical gradients that uses Pluronic F-127, a synthetic block copolymer that, as a 23% aqueous solution, forms a liquid at low temperature and a gel at room temperature. Test chemicals are put into a modified pipette tip, or 'chemical dispenser,' and dispensers are inserted into the gel in which nematodes have been dispersed. Meloidogyne hapla is attracted to pH gradients formed by acetic acid and several other Brønsted acids and aggregates between pH 4.5 and 5.4. While this pH range was attractive to all tested root-knot nematode strains and species, the level of aggregation depended on the species/strain assessed. For actively growing roots, the pH at the root surface is most acidic at the zone of elongation. This observation is consistent with the idea that low pH is an attractant for nematodes. Root-knot nematodes have been reported to be attracted to carbon dioxide, but our experiments suggest that the observed attraction may be due to acidification of solutions by dissolved CO₂ rather than to CO₂ itself. These results suggest that Pluronic F-127 gel will be broadly applicable for examining responses of a range of organisms to chemical gradients or to each other.

Keywords Nematode · Chemotaxis · Carbon dioxide · Gradient · Meloidogyne hapla · Meloidogyne incognita · Meloidogyne javanica · Bioassay

Introduction

Root-knot nematodes (RKNs) (Meloidogyne spp.) are destructive pathogens of agricultural crops. The infective stage, the second stage juvenile (J2), is attracted to host roots, especially the zone of elongation of growing roots, and migrates by sinusoidal movements toward roots, presumably due to a concentration gradient of a substance or substances from the root (Linford 1939; Wyss et al. 1992; Wang et al. 2009). Disrupting host-finding behavior may be an effective way of controlling plant-parasitic nematodes. However, little is known about the chemical signals that the nematode perceives from the root. Several assays have been used to study chemotaxis in nematodes, but all have limitations (Spence et al. 2008). Many assays have used agarose plates (e.g., Ward 1973; Riddle and Bird 1985); however, nematodes move along the surface of the agar in a film of water, and thus, the assay is 2-dimensional. Nematodes move on their sides on agar surfaces and perceive chemicals through amphids (anterior chemoreceptors). They detect chemical gradients by side-to-side (up and down on agar surfaces) displacement of the amphids due to head movements (Ward 1973). Other attraction assays have used sand-filled tubes in which movement of nematodes cannot be observed directly (Robinson 1995).

Arguably the most frequently suggested attractant for plant parasitic nematodes has been carbon dioxide (Klingler 1965; Prot 1980; Pline and Dusenbery 1987; Robinson 1995). Perry (2005) suggested that CO₂ was a long distance attractant, and that local attraction was due to gradients of
other signals such as pH. However, the diversity of sources of CO₂ in the rhizosphere and the imperfect correlation between regions of the root with high CO₂ and those with the highest attraction suggest that CO₂ may not be a specific signal (Bird 1960). In the soil, CO₂ efflux is the result of root and rhizospheric respiration and organic matter decomposition. Meloidogyne incognita is repelled by ammonia and several nitrogenous salts released from decaying material (Castro et al. 1991) possibly further modulating its response. Several nitrogenous salts released from decaying material (Sternberg 2008). Recent studies with Caenorhabditis elegans, CO₂ has been described both as an attractant and a repellant (Dusenbery 1983; Bretscher et al. 2008; Hallem and Sternberg 2008). Recent studies with C. elegans suggest that the degree of CO₂ avoidance depends on strain and nutritional status (Hallem and Sternberg 2008).

Pluronic F-127 (PF-127) has been used by our group to study plant-nematode interactions and nematode behavior (Wang et al. 2009). A 23% PF-127 solution forms a semisolid gel at room temperature but is liquid at 15°C and below. The gel is highly transparent and generally has low toxicity or is non-toxic, depending on the organism and gel concentration used (Gardener and Jones 1984; Ko and Van Gundy 1988). Nematodes can be mixed in PF-127 solution at 15°C. As the solution warms to room temperature, nematodes become suspended in the gel. They can move freely in 3 dimensions through the gel, allowing examination of behavior in response to various treatments. Nematodes dispersed in PF-127 gel are highly attracted to roots, aggregating mainly at the zone of elongation (Wang et al. 2009). In the absence of roots, nematodes aggregate into tight balls containing most of the nematodes that were distributed in PF-127 gel. We have postulated that this aggregation is due to a preference for low oxygen concentration, a response to anhydrobiotic stress, and/or inter-nematode chemical signals (pheromones). The ability to form aggregates differs among RKN species and strains (Wang et al. 2009; Wang and Williamson, unpublished).

Here, we tested the utility of PF-127 gels for examining root-knot nematode movement in gradients of pH and CO₂. Because the genetic map and genome sequence for Meloidogyne hapla strain VW9 are available (Opperman et al. 2008), thus opening the potential to explore genes involved in signal recognition and behavior by the nematode, we focused on this nematode in the current work. We have, however, compared the behavior to isolates of other agronomically important species, including M. incognita, for which the genome sequence is also available (Abad et al. 2008).

**Methods and Materials**

Nematodes and Materials Meloidogyne hapla strains used were VW9 (Liu and Williamson 2006), NCS (from C. Opperman, North Carolina State University) and LM (from P. Roberts, University of California, Riverside; Chen and Roberts 2003). M. javanica strains VW4 and VW5 are a pair of nearly isogenic strains that differ in ability to reproduce on tomato bearing the resistance gene Mi-1 (Gleason et al. 2008). M. incognita strains used were 557R (Yaghoobi et al. 2005), W1 (inbred from nematodes isolated from a Woodland, CA tomato field), Beltran (P. Roberts), Harmony (H. Ferris, University of California, Davis), and VW6 (inbred from a field isolate obtained by H. Ferris). All M. hapla strains, M. javanica strain VW5, and M. incognita strains 557R and W1 were maintained on tomato cultivar VFNT, which carries the gene Mi-I. The remaining strains were maintained on the nematode-susceptible tomato cultivar UC82. Eggs were collected and axenic J2s produced as described by Branch et al. (2004) with minor modifications (Wang et al. 2009), then rinsed four times with sterile water; larvae hatched at room temperature.

Methanesulfonic acid and pyridine, both 99%, were from Acros Organics (Geel, Belgium). Bromocresol purple was the sodium salt from National Analyne Division, New York, NY, U.S.A.. Other chemicals were A.C.S. grade.

**Gel Preparation and Attraction Assay** PF-127 (Sigma, St. Louis, MO, U.S.A.) solution was 23% w/v in 10 mM Tris, 10 mM morpholino-ethanesulfonic acid (MES), pH 7.1, unless otherwise indicated. Suspended PF-127 was dissolved by stirring for 24 h at 4°C (Wang et al. 2009). Where noted, the pH indicator bromocresol purple was included. The solution was stored at 15°C. Twenty ml of PF-127 solution containing freshly-hatched J2 (6000 J2 per plate unless otherwise indicated) were poured into each Petri dish (nominal 100×15 mm) at 15°C. “Chemical dispensers” were prepared by cutting standard 5 cm, 200µl pipette tips (USA Scientific Inc.) 5 mm from the small end and 20 mm from the large end. The test chemical solutions were prepared in 23% PF-127 and maintained on ice. Approximately 100µl of the test solution were pipetted into each chemical dispenser, which was held horizontally while the solution was allowed to gel at room temperature. The Petri dish with the PF-127 solution and dispersed nematodes was placed at room temperature, and two dispensers were placed anti-parallel with ends not closer than 3 cm from the Petri plate edge, before the gel formed and before bubbles could develop at the dispenser ends. Dispensers rested on the bottom of the plate, and the top of the large opening of the dispenser was approximately at the surface of the gel in the Petri dish when 20 ml of PF-127 solution were used (i.e., 3.4 mm deep gel). At the indicated time after initiation of the assay, the number of nematodes was counted within the 5 mm diam. circle centered at the small end of pipette tip. Control dispensers containing 23% PF-127 in water were used as checks.
different sizes of the two dispenser openings allowed us to examine two different concentration gradients simultaneously, and the paired dispensers in each plate provided duplicate analyses. At least three plates were included in each experiment, and each experiment was repeated at least twice. Micrographs were captured with a Nikon SMZ-U dissecting microscope by using SimplePCI High Performance Imaging System (Compix Inc, Sewickley, PA, USA). To obtain a broader view of the nematode distribution patterns in response to chemical gradients, lower magnification photographs of Petri plates were taken with a digital camera by using edge lighting and against a black background.

**pH Measurements** The pH in gels was measured with a pH Meter (Denver Instrument Company, Arvada, CO, USA), by direct insertion into gel of a PHR-146 Micro Combination pH electrode (Lazar Research Laboratories, Inc., Los Angeles, CA, USA). pH meter calibration was performed by the two point standardization method (pH 7 and pH 4) according to the manufacturer’s instructions. pH measurements were repeated at least twice.

**Carbon Dioxide Delivery Assay** A Petri dish lid was modified for delivery of CO2 to PF-127 gel (Fig. 1). CO2 sublimed from dry ice in a container was passed via tubing through water and delivered to a central 1.5 cm diam. circle on the gel surface. Water-saturated air was delivered to the remainder of the gel (Fig. 1), thus creating a concentric gradient of CO2 in air in the gel from the 1.5 cm circle to the edge of the plate.

**Statistical Analysis** JMP software (SAS Institute Inc., Cary, NC, USA) was used for statistical analyses. Data were subjected to one-way analysis of variance (one way–ANOVA). Results are reported as significant or non-significant in Tukey’s Honestly Significant Difference (Tukey HSD) Test ($P<0.05$).

**Results**

**Formation of pH Gradients in Gel** To assess gradient formation in PF-127 gel, the pH indicator bromocresol purple was incorporated into gel in 10 mM Tris-MES buffer, pH 7.1. Two dispensers with 0.17 M acetic acid were placed into each plate. After 4 h, yellow circles were seen around the openings of the dispenser, indicating that proton diffusion occurred in a radial pattern from the openings of the dispenser. The size of the circles increased slowly over the next 20 h (Fig. 2).

**Aggregation of M. hapla (Strain VW9) in pH Gradients** The response of *M. hapla* strain VW9 J2 was monitored in pH gradients produced around the ends of dispensers that had been charged with 0.17, 0.34, or 0.85 M acetic acid. Nematode clustering was observed at the dispenser small end 3 h (not shown) after insertion and was maintained after 5 and 24 h (Fig. 3A and B, respectively). Five h after insertion, the gradient near the dispenser large end extended over a sufficient gel volume so that J2 responded by congregating into a halo (Fig. 3C, 0.34 M, and 0.85 M acetic acid dispensers). Presumably, the halo was generated because J2 prefer a specific low pH range to the even lower pH near the dispenser end and the pH 7.1 of the PF-127 gel in which J2 were originally suspended. Figure 4 presents J2 density related to distance from the dispenser large end for 0.34 M and 1.7 M acetic acid. The observed distributions conformed to a normal curve, and the gradient created by 0.34 M acetic acid stimulated J2 to achieve a density ten times greater than the initial density of J2 in the gel at the most dense part of the halo (Fig. 4).

By using a microelectrode, pH was measured at the boundaries of the halo of high nematode density. For both low pH and high pH boundaries, more than 50 readings were made from three plates. At 7 h after assay initiation, the pH was 4.54±0.05 for the inner boundary and 5.40±0.15 for the outer boundary. At 20 h after assay initiation, the circle was larger, but the pH values at the boundaries were similar (pH 4.58±0.17 for inner boundary and pH 5.37±0.13 for outer boundary). The experiment was repeated, and the same boundary pH values were obtained. No aggregation of nematodes was observed with any of the checks that contained dispensers with 23% PF-127 in water (not shown).

To test whether acetate from ionized acetic acid was sufficient to account for gradient-directed nematode migration, the response of *M. hapla* J2 was observed for dispensers with 10 and 50 mM sodium acetate, corresponding to the range of acetate ion expected in the previously tested acetic acid-loaded dispensers. At 24 h, no nematode aggregation was seen at the openings of the dispensers (results not shown), suggesting that protons or
acetic acid rather than acetate ions affected the nematode aggregation.

By using the dispenser system, we tested the response of VW9 to a variety of Brønsted (proton-donating) acids, including strong acids (HCl, H$_2$SO$_4$, HClO$_4$, methanesulfonic acid), carboxylic acids (acetic, citric, formic, lactic, propionic, and succinic), pyridinium hydrochloride, and pyridinium methanesulfonate. Strong acids and three
monoprotic carboxylic acids (acetic, formic, and propionic acids) induced movement of the nematodes to form a halo centered at the wide opening of the dispenser (Figs. 3C and 5A–C) and aggregation at the small opening (Figs. 3A&B, and 5G). For lactic, citric, and succinic acids, the J2 migrated to and aggregated inside the dispensers (for example, see Fig. 5E and H). The ability of citric acid and citrates to act as a J2 attractant was tested by charging dispensers with 0.40 M citrate plus citric acid buffers at pH~4.5 or ~5.5 (corresponding the observed pH values at the halo edges) and with sodium, potassium, ethanalammonium, or diethanolammonium as the cation. All of these citrate buffers, like citric acid itself, attracted J2 well into the dispensers, rather than just to the openings, suggesting that citrate/citric acid attracted J2 sufficiently to overcome any tendency to form a halo in a citric acid gradient (results not shown). Pyridinium ions were selected as another Brønsted acid because of the similarity of the pKₐ (5.2) to that of acetic acid (4.8). Pyridinium methosulfonate (Fig. 5E) and pyridinium chloride (not shown) also caused aggregation of J2 inside the tip in a manner similar to that seen for citric acid.

Response of Nematodes to Gel Exposed to CO₂
To investigate RKN J2 response to CO₂, we delivered CO₂ to a 1.5 cm circle in the center of a PF-127 gel plate buffered with 10 mM Tris-MES, pH 7.1, in which J2 of VW9 were distributed evenly. In a parallel experiment, CO₂ was delivered to a plate with the same gel, but without nematodes and with the pH indicator bromocresol purple. By two hours, the region of the gel exposed to CO₂ had aggregated much more slowly than the other strains. By 24 h and for several species and strains, the number of nematodes near the dispenser tip was too great to count with confidence, and was scored as described in Fig. 6. However, even for strain Harmony, the number of nematodes at the tip increased between 3 and 24 h, and the relative aggregation of the different strains remained the same (Table 1). The pH range of maximum aggregation was assessed at 24 h by using a microelectrode, and all were found to aggregate between pH 4.5 and 5.4. Thus, it did not appear that the strains differed in the pH range to which they were attracted, but they did differ in the rate at which they congregated in the gel region of that pH range.

Root-knot Nematode Species and/or Strains Differ in Responsiveness to Acetic Acid
Different species and strains of root-knot nematodes differ in the rate at which they aggregated at root tips in PF-127 gel (Wang et al. 2009). Responses of different RKN species and strains to acetic acid gradients were compared 3 h after initiation of the assay (Table 1). Meloidogyne hapla strain VW9 had the highest aggregation at the dispenser tip of all the RKN strains tested. The nearly isogenic M. javanica strains VW4 and VW5, which differ in their ability to parasitize tomato with the resistance gene Mi-1, did not show significant differences in attraction to acetic acid. However, significant differences were seen among M. incognita strains derived from isolates originating from different locations and different hosts. Strains VW6, Beltran and W1 aggregated significantly less than M. incognita strain 577R. M. incognita strain Harmony, which was originally isolated from grape, aggregated much more slowly than the other strains. By 24 h and for several species and strains, the number of nematodes near the dispenser tip was too great to count with confidence, and was scored as described in Fig. 6. However, even for strain Harmony, the number of nematodes at the tip increased between 3 and 24 h, and the relative aggregation of the different strains remained the same (Table 1). The pH range of maximum aggregation was assessed at 24 h by using a microelectrode, and all were found to aggregate between pH 4.5 and 5.4. Thus, it did not appear that the strains differed in the pH range to which they were attracted, but they did differ in the rate at which they congregated in the gel region of that pH range.

Fig. 4 Distribution of VW9 J2 in acetic acid gradients. Dispensers containing acetic acid in 23% PF-127 gel were immersed in buffered PF-127 gel containing 0.30 J2 per mm³. After 5 h, the gel was photographed with edge lighting. Ninety degree arcs, scaled to 0.5 mm wide in the original gel, were superimposed on the enlarged photographic images with the origin of the arcs at the middle of the dispenser wide end. J2 within each arc were counted for the 0.34 M (●) and 1.7 M (○) acetic acid dispensers (four dispensers from three plates), and the data were transformed to J2 per mm³ of gel. The average values and standard deviations are presented against distance from the dispenser large end. The standard deviation expanded at small distances from the dispenser because of the small volumes corresponding to the short arcs. Data were fit to Gaussian curves by a least-squares method.
was delivered, then moved inside during the next 24 h after the source of CO$_2$ was disconnected. At the latter time the pH inside the circle was ~5.8 whereas that outside was ~6.1.

To distinguish the response to CO$_2$ from that of reduced pH, the CO$_2$ experiment was repeated with gel buffered with 10 mM sodium acetate, pH 4.9. At 24 h after continuous CO$_2$ exposure, nematodes remained uniformly distributed throughout the gel and did not aggregate at the border of the circle (Fig. 7B). Using a microelectrode, the pH inside the circle was determined to be approximately 4.7 and that in the surrounding gel 4.9. At 24 h after removal of the CO$_2$ source, nematodes aggregated within the circle but to a much lesser extent than when the gel was buffered at pH 7.0.

**Discussion**

All tested strains of RKN J2 aggregated in a gel zone with a pH range of 4.5–5.4, e.g., from an acetic acid gradient. These species, as well as many other plant pathogens/symbionts, are strongly attracted to and invade roots in the zone of elongation. Growing root cells extrude protons, with the elongation zone being the most acidic region (Mulkey and Evans 1981; Pilet et
Based on pH indicator dyes, this zone was found to be less than pH 5 for maize seedlings grown on agar plates (Mulkey and Evans 1981). Using PF-127 gel and pH indicator dyes, we also found that *Medicago truncatula* and tomato seedlings produced an acid gradient around the root and could acidify the medium to pH 5 or less immediately adjacent to the root (Wang and Williamson, unpublished). The correlation of the most acidic region of the root with that which is most attractive to the nematodes is consistent with the possibility that pH gradients are signals that direct nematodes to the elongation zone of growing roots. However, while low pH is a marker for the growth zone of a wide range of host plants, the influence of the growth zone on pH in the soil is likely much less than it is in water or gels (Nichol and Silk 2001). In addition, we found that while the preferred pH range was the same for all root-knot nematode species tested, the rate of aggregation within the zone of preferred pH varied with the species and strains of root-knot nematode. *Meloidogyne hapla* VW9 aggregated most rapidly, whereas all *M. incognita* strains, except 577R, were significantly slower to aggregate than *M. javanica* and *M. hapla* species. These results cannot be reconciled directly with our previous observations of the relative migration rates of these same nematode species to host roots; i.e., movement of *M. incognita* and *M. javanica* J2 to tomato roots was much more rapid than that of *M. hapla* (Wang et al. 2009). Thus, pH is probably not the sole attractant to roots, and its relative importance may depend on the nematode strain as well as the host.

### Table 1 Attraction to acetic acid of species and strains of root-knot nematode

| Species/Strain | Isolate host and location | Nematodes within 5 mm of tip |
|----------------|--------------------------|------------------------------|
| **Meloidogyne hapla** |                          |                              |
| VW9            | Tomato, CA               | 106.8A                       |
| NCS            | Unknown, NC              | 62.8B                        |
| LM             | Unknown, France          | 71.5B                        |
| **M. javanica** |                          |                              |
| VW4            | Unknown, CA              | 57.0BC                       |
| VW5            | Unknown, CA              | 50.5C                        |
| **M. incognita** |                          |                              |
| VW6            | Cotton, CA               | 26.8D                        |
| Vella          | Lima bean, CA            | 37.0D                        |
| Harmony        | Grape, CA                | 4.5E                         |
| 557R           | Tomato, NC               | 58.8BC                       |
| W1             | Tomato, CA               | 28.8D                        |

*a Two chemical dispensers filled with 0.17 M acetic acid were used for each plate and three plates for each treatment. At the indicated time after insertion of the dispensers, the number of nematodes within the 5 mm diam. circle, centered at the small end of pipette tip, was counted. The counts were corrected by subtracting similarly counted nematode numbers for dispensers containing gel alone. Significant differences between counts on different nematode strains at the 3 h time point are indicated with letters A-E (*P*< 0.05). Nematodes continued to aggregate so that counts were not possible at 24 h for some trials, but relative nematode densities were assigned symbols +, ++, and +++ following the examples in Fig. 6. For the four strains whose numbers were assessed at the 24 h time point, significant differences are indicated by letters A and B (*P*< 0.05).

**Fig. 6** Scoring aggregation of root-knot nematodes to acetic acid. The small opening of the dispenser with 170 mM acetic acid is shown at 24 h after initiation of the assay. **A.** *M. javanica* strain VW4. This sample would be scored as “+” in Table 1. **B.** *M. hapla* strain LM. This sample was scored as “++.” **C.** *M. hapla* VW9 was scored as “+++.” Scale bar = 1 mm.
While all the acid solutions that we tested resulted in nematode aggregation, some appeared to be more attractive than others. For example, for citric, lactic, and succinic acids, nematodes migrated into the dispensers rather than forming a halo around the tip, suggesting that these acids may be specific attractants. In support of this, citrate salts with various cations at pH values of 4.5 or 5.5 also caused the nematodes to aggregate within the tip. Pyridinium hydrochloride and pyridinium methane sulfonate were tested because these Brønsted acids are similar to the tested carboxy acids in their acid gradient forming potential. However, the pyridinium salts proved to be strong attractants to RKN J2s; in agreement with this observation, pyridine has been reported to be an attractant for *C. elegans* (Dusenbery 1976). It is likely that nematodes respond to and integrate complex signals and/or combinations of signals in their natural environment and that perceived gradients may act in a synergistic or antagonistic manner. The assay described here has potential to identify and dissect some of these interactions.

Carbon dioxide has been widely reported to be an attractant to free-living, insect-parasitic and plant-parasitic nematodes (Pline and Dusenbery 1987; Robinson 1995; O’Halloran and Burnell 2003). However, some of these studies were carried out without buffers, and carbon dioxide can lower pH of aqueous solutions. In our assays, nematodes were attracted to just outside the CO2-permeated central cylinder (Fig. 1) when the gel was in 10 mM Tris-MES, pH 7.1 (Fig. 7A), whereas nematodes remained uniformly distributed throughout when the gel was buffered with 10 mM sodium acetate, pH 4.9 (Fig. 7B). A pH value of 4.9 was selected because it is within the pH 4.5 to 5.4 range observed for the halos of nematodes that formed in acetic acid gradients (Fig. 3C) and because the calculated pH for water saturated with CO2 is 4.4. That is, when the initial pH is 4.9, little perturbation of the gel pH is expected from introduction of CO2; our gel pH measurements agreed with this expectation. Thus, for the pH 7.0 gel assay, it is likely that what attracted J2 to the region around the central cylinder was a pH gradient, not CO2.

For either pH 7.0 or 4.9, the number of nematodes in the central cylinder of gel did not change in number while the cylinder was continuously exposed to CO2. However, nematodes entered this area after the delivery of CO2 was terminated. There are several possible explanations for the failure of the nematodes to enter the CO2-exposed region. CO2 is an anesthetic, and concentrations of CO2 above 10% have been reported to cause a decrease in movement for *M. incognita* (Pline and Dusenbery 1987). Thus, perhaps the nematodes regained ability to move into the central region once the level of the dissolved gas had decreased sufficiently. Our pH measurements indicate that the pH remained lower in the central region compared to the periphery, possibly contributing to the attraction. In addition, it is likely that the oxygen level was lower in the central region compared to the periphery, possibly contributing to the attraction. In addition, it is likely that the oxygen level was lower in the region exposed to CO2, and root knot nematodes may prefer less than ambient oxygen levels as does *C. elegans* (Gray et al. 2004). As with any in vitro assay, caution should be used in extrapolating in vitro experiments with more complex soil conditions. However, this system has the potential to provide new insights into nematode behavior in natural soil.

The method of dispenser immersion in PF-127 gel described here is simple in design and execution, and it produced steady-state pH gradients in which nematode movement was easy to visualize and document over time. Results were reproducible enough in both nematode distribution and maximum density, so that data from several plates could be combined and presented as a close approximation to a normal distribution of J2 density vs. distance from the dispenser end (Fig. 4). This system has advantages over agarose-based methods in which the test chemical is applied to a well in the agar (Wuyts et al. 2006) or on the surface of the gel (Ward 1973; Matsuura et al. 2007) to form a gradient. For agar-based assays, the volume of test substance that can be applied is limited, and sometimes a second application is needed to maintain the
gradient, whereas with our system the dispensers provided an ample reservoir of test chemical. The PF-127 assay medium in which nematodes are dispersed is three-dimensional. Therefore, the nematode is not constrained to orientations that may be sub-optimal for perception of chemical gradients by amphids and can move in any direction in response to a detected gradient. Photographs with edge lighting allowed population scale analyses of nematode behavior in response to chemical stimuli, although nematode entry into the dispensers was obscured. Under the microscope, local gatherings of nematodes, in the dispenser and elsewhere, were observed, and the clarity of the PF-127 gel allowed the examination of the behavior of individual nematodes.

Assays of this type have great potential for chemotaxis-response studies with a variety of potentially attracting substances, including root tissues (Wang et al. 2009) or small organisms, and should be suited for the study of the responses of a variety of small invertebrates, fungal zoospores, and other subjects.

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