Structural Transformations of Low-Temperature Quartz During Mechanoactivation

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Abstract. The paper presents the study of mechanoactivation impact on crystal structure of α-quartz. The volume of silicon-oxygen tetrahedron SiO$_4^{4-}$ is accepted as the structural parameter depending on the mechanoactivation degree. The paper compares the dependence of this parameter on temperature, pressure and time of mechanoactivation of α-quartz in a planetary mill.

Keywords: Quartz · Crystal structure · Silicon-oxygen tetrahedron · Mechanoactivation

1 Introduction

The mechanoactivation dispergation of quartz materials is a widespread method of technological processing of this mineral raw material in various fields of technological mineralogy.

In practice of technical petrogenesis of cast stone, forming the basis of synthesis of inorganic silica-containing binding agents (Dmitrieva et al. 2018), the mechanoactivation dispergation of quartz raw material holds a special place.

The result of almost a century-long study of phase and structural transformations of low-temperature quartz during mechanoactivation is the amorphicity of a surface layer of quartz particles and the formation of nanosized β-quartz crystals in α-quartz matrix (Zhernovsky et al. 2018). At the same time there was no study on structural transformations of α-quartz within a quartz particle matrix during mechanoactivation. The only exception is the study by Archipenko et al. (1987, 1990) concerning phase transformations in mechanoactivated α-quartz. The task of this study is to fill the gap in this matter partially.

2 Materials and Methods

The grinding of hydrothermal quartz (Ural) was carried out in the PULVERISETTE 6 classic line planetary mill (Fritsch, Germany) with lining and grinding bodies of tungsten carbide. The milling time made 3, 12, 30, 60, 120, 180, 240, 300 and 360 min.
The diffraction spectra of samples are obtained using ARL 9900 Workstation (λCo). Shooting interval – 2θ:8–80, step angle – 0.02. The specification of structural parameters was carried out in DDM v.1.95e software for the difference curve derivative (Solovyov 2004). 174-ICSD data (P3_21) were used as a structural model. Profile function – pseudo-Voigt. The specification of the profile parameters was carried out in the anisotropic approximation. Thermal corrections were specified in anisotropic option. Table 1 shows the experimental results.

| Milling time, min | 3   | 12  | 30  | 60  | 120 | 180 | 240 | 300 | 360 |
|-------------------|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| Si x/a            | 0.4499 | 0.463 | 0.460 | 0.466 | 0.465 | 0.466 | 0.468 | 0.469 | 0.470 |
| O x/a             | 0.403 | 0.416 | 0.413 | 0.419 | 0.418 | 0.419 | 0.422 | 0.422 | 0.423 |
| O y/b             | 0.2660 | 0.2702 | 0.2697 | 0.2715 | 0.2668 | 0.272 | 0.2697 | 0.2721 | 0.2707 |
| O z/c             | 0.7950 | 0.7921 | 0.7919 | 0.7905 | 0.7886 | 0.7868 | 0.7866 | 0.7887 | 0.7876 |

Note: Si y/a = 0, Si z/c = 2/3.

### 3 Results and Discussion

The volume of silicon-oxygen tetrahedrons was chosen as a structural and sensitive quartz parameter on mechanoactivation influence since it depended on several elementary structural parameters: bond length Si-O, bond angle O-Si-O and twist angle of tetrahedrons (Goryaynov and Ovsyuk 1999).

It is known that mechanoactivation is the result of two processes influencing the material – local thermal influence and impact pressure. Hence, it is advisable to consider structural changes of quartz during mechanoactivation in comparison with changes during thermal and baric impacts. The following were used as references on changes of quartz structural parameters: at a thermal influence – the work of Kihara (1990), at a baric impact – works of Levien et al. (1980), Hazen et al. (1989) and Glinnemann et al. (1992).

The given results show that mechanoactivation alongside with thermal and baric influences transforms the structure of α-quartz by reducing the volume of SiO_T-tetrahedrons. At the same time the dependences of volumes of α-quartz silicon-oxygen tetrahedrons on severity of exposure differ a lot (Figs. 1 and 2).

The energy accumulation by quartz during mechanoactivation, as well as under thermal and baric influence most likely happens due to the reduction of Si-O bond length.

The energy accumulated by quartz during mechanoactivation can be estimated by comparing the dependences (Figs. 1a and 2). This comparison confirms that material energy saturation after three-hour mechanoactivation is equivalent to its heating up to 500 °C.
Conclusions

The change of volume of silicon-oxygen tetrahedrons may be considered as an indicator of the α-quartz crystal structure response to the thermal and baric influence, as well as to the mechanoactivation. The difference of these mechanisms of influence is demonstrated by different dependences of this parameter of α-quartz crystal structure on the severity of exposure.

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