Manganese-doped cerium oxide nanocomposite as a therapeutic agent for MCF-7 adenocarcinoma cell line

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**ABSTRACT**

The preparation of a manganese-doped cerium oxide (Mn:CeO$_2$) nanocomposite via hydrothermal route is described. Cubic fluorite structure of single phase was exhibited by studying structural analysis through x-ray diffraction (XRD) technique and morphological analysis was conducted by scanning electron microscope. Surface analytic technique of energy dispersive x-ray spectroscopy (EDX) was conducted to analyze the relative amount of any impurity and doping. Structural changes due to manganese doping such as increment in production of vacancies of oxygen within crystal of cerium oxide, and reduction in size of crystallite and constant of lattice was observed in our research study. Moreover, the Mn:CeO$_2$ nanocomposite demonstrates differential cytotoxicity against MCF-7 adenocarcinoma cell line, which renders it a promising candidate for targeted cancer therapy. The anti-tumorous activity of the cerium oxide nanocomposite was significantly enhanced with doping of manganese, which is directly linked with the generation of highly reactive oxygen facets. The experimental results are supported by a mathematical model that confirms a confidence level of 95%. This research has paved the way for many utilities in therapeutics and magnetic resonance imaging diagnostics through new observations, and hence verified their math model.

1. Introduction

The cytotoxicity of metallic nanoparticles (NPs) is known to be enhanced by doping with other materials (Wojcieszak et al., 2018). Doping often results in an enhancement of the production of reactive oxygen species (ROS), which renders these nanoscale particles associated with metals or metal oxides good candidates for treatment of different cancer and bacterial diseases (Luo et al., 2018). Cerium oxide (CeO$_2$), also called ceria, is an insulating and ionic oxide with a cubic fluorite-like structure (CaF$_2$) containing Ce$^{3+}$ and Ce$^{4+}$ ions. Although ceria itself has no particular biological significance (Kumar et al., 2013); its nanoparticles having the ionic form Ce$^{3+}$ are relevant biocompatible ingredients of antiemetic, bacteriostatic, bactericidal, and antitumor agents (Abbas et al., 2015). In addition, due to their reactive behavior, these NPs are suitable for treatment of fetal diseases like cancer, and their ROS activities have been recently a research target (Saranya et al., 2014). The biological properties of these NPs, including biodistribution, toxicity, elimination, and dissolution, depend on their physical and chemical characteristics.
CeO₂ also exhibits catalytic properties that stem from its redox properties, structural defects such as high reactive oxygen species and their harmful effects during catalytic process (Asati et al., 2010). These catalytic properties, as well as the oxygen reserve and transfer capacity of CeO₂, can be enhanced by doping, which has led to the development of biomedical applications (Chaudhari et al., 2017).

Manganese, which has an abundance of 0.1% and is one of the 12 most occurring metals in the earth crust, is one of the most promising dopants for CeO₂ NPs because its ionic radius is similar to that of Ce (Parvathya and Venkatramanb, 2017). An essential nutrient for animals and plants with a key role in bone formation and the urea cycle, Mn is usually found in nature forming oxides, carbonates, and silicates in more than 100 minerals. Its wound healing ability has attracted the attention of researchers, and its toxicological properties, along with that of other metallic nanoparticles, is a hot research topic.

For cancer treatment, ROS production in malignant cells is actively investigated as one of the major factors for destruction of infected cells. This production of ROS is highly dependent on defects states in cells. In CeO₂ NPs, defects may be generated by Mn doping; therefore, this can be envisaged as an effective approach to enhance the anticancer activity in cells (Ansari et al., 2016). Herein, CeO₂ NPs are doped with Mn to examine its toxicological effects (Celardo et al., 2011). Considering the need for a cost effective and environmental friendly preparation method that allows morphology control (Jayakumar et al., 2017), the present Mn-doped CeO₂ nanocomposites are prepared via hydrothermal approach, which is safe, can be conducted at room temperature, and enables NP size control by controlling the pH. The successful growth and structure of these nanoparticles is confirmed by diffraction technique: XRD and electronic microscopic scan: SEM. The bacterial and photodynamic effects on an MCF-7 adenocarcinoma cell line are investigated. A schematic illustration of the preparation process is shown in Fig. 1.

2. Materials and methods

2.1. Chemical and reagents

Ceirum (III) nitrate hexahydrate (Ce(NO₃)₂·6H₂O; 432.2 g/mole; 99.9% purity), manganese chloride (MnCl₂·4H₂O), sodium hydroxide (NaOH; 40 g/mole; 99.9% purity), and acetic acid (CH₃COOH) were procured from Sigma-Aldrich Chemicals. Double distilled water was used as solvent to prepare the solutions. In all the experimental work, acid washed glassware was used.

2.2. Hydrothermal synthesis of manganese-doped cerium oxide nanoparticles

Mn-doped CeO₂ NPs were synthesized by hydrothermal method using MnCl₂·4H₂O and Ce(NO₃)₂·6H₂O as precursors. In a typical procedure, a 1 M solution of Ce(NO₃)₂·6H₂O in 100 mL of deionized water was mixed with a 7 M solution of MnCl₂·4H₂O under constant magnetic stirring with a speed of 7000–10000 rpm. Then, 2 mL of acetic acid was poured drop by drop into the solution as a capping agent to control/minimize the particle size of the nanocomposite, and a basic solution of 1 M (8 g/100 mL) NaOH was slowly added under constant stirring to attain a pH of 10. After 1–2 h, a brownish white colloidal solution was obtained. Then, the colloid was separated by centrifugation, and the precipitate was placed in a water bath at 75 °C for 20–30 min. The collected sample was shifted to the Teflon auto-clave. The dried sample was then calcined in an electric furnace at 600 °C for 4–6 h to obtain the final NPs, which were subjected to XRD analysis (Jayakumar et al., 2017; Syed Khadar et al., 2019).

2.3. Anticancer activity and cytotoxicity analysis

The percent cell viability and anticancer activity of the Mn:CeO₂ nanocomposite in the absence or presence of 5-ALA drug was assessed for MCF-7 breast cancer cell line through colorimetric assay for assessing cell metabolic activity. For this experiment, the chosen cell line was collected at an oncology lab of a tertiary care hospital in Islamabad. Cytotoxicity and analysis of generation of reactive oxygen species: ROS analysis of the manufactured nanocomposites toward these tumorous cells were performed through our published protocol in reported articles (Atif et al., 2019; Ahamed et al., 2016; Mehmood Ur Rehman et al., 2017). Cells were grown in RPMI complemented with 10% fetal calf serum (FCS) and 5% GPPS antibiotics in a 75 cm² culture flask at 37 °C in the presence of 5% CO₂ in a SHELLAB incubator. The cell lines were grown in a cultural plate:96-well plate having proportion of 10⁵ cells per well. The cells were admitted to stick to the surfical wall for 48 h by incubating them at 37 °C and 5% CO₂ for one day. After which, the cells were treated with Mn:CeO₂ nanocomposite (20 g/mL) for one day. The suspensions of the Mn:CeO₂ nanocomposite were poured to the cells. Cells in absence of nanocomposite were taken as control in this conducted study. They were poured into the cultural plates at a concentration of 10⁴ cells/well along with various concentrations of Mn: CeO₂ nanocomposite and further incubated for a day. Briefly, 10 μL MTT (5 mg/mL) was poured to the wells for determining viability of the cells. The cultural plate was further put in the incubator for 2–3 h under the same incubation conditions. Afterwards, a solubilization process, the
Absorbance was recorded at 550 nm using a PLATOS R 496 plate reader (Atif et al., 2019; Iqbal et al., 2015).

Formulation of viable cells was measured by:

\[
\text{Percentage of viable cells} = \frac{\text{Absorbance of the nanocomposite sample} - \text{Absorbance of the Blank}}{\text{Absorbance of --ve control} - \text{Absorbance of Blank}} \times 100
\]

2.4. Characterization

The crystallographic study of the manufactured nanocomposite was conducted using a D8 Advance X-ray diffractometer (Bruker AXS) using monochromatic Cu Kα rays having 2 0 speed of scan per minute. The dimensions (size) and surficial study was measured and analyzed by SEM (Tescan Vega 3 LMU SEM). The absorbance of the bacterial cultures was observed using a nanodrop Thermo 2000 °C spectrophotometer (Thermo Fisher Scientific, USA) with a peak analyzer system. A Platos R 496 microplate reader (Ameda Labor diagnostik GmbH, Graz, Austria) was used to measure the absorbance of the MTT-treated cell line at 550 nm.

3. Results

3.1. Structural study

Typical XRD patterns of the Mn: CeO₂ sample calcined at 600 °C are depicted in Fig. 2. The obtained 2θ values of 28.6°, 33.0°, 47.5°, 56.4°, 59.3°, and 69.5° are attributable to the (111), (200), (220), (311), (222), and (400) planes, respectively, similarly as described in previous reports (Saikia et al., 2017). The ionic radius of Ce⁴⁺ (1.03 Å) is larger than that of Mn²⁺ (0.74 Å) and, therefore, the latter ions can substitute the former in the CeO₂ crystal lattice. The crystallographic variables, i.e., D; size of the crystallite and constants of lattice (a = b = c) were measured using the given formulation:

\[
D = \frac{0.89 \lambda}{\beta \cos \theta} \quad \text{and} \quad d = \frac{a = b = c}{\sqrt{h^2 + k^2 + l^2}}
\]

2. Percentage of viable cells calculated mean size of the grain was average NPs 105 nm.

3.2. Morphology analysis and energy dispersive X-Ray (EDX) analysis

The study of anatomies of the manufactured nanocomposite were analyzed by SEM. Fig. 3(a) depicts the structural anatomy of the Mn: CeO₂ sample. The SEM micrographs clearly reveal the spherical structure in nano-range, which appeared to have homogenous distribution of shape and size of the particle. The calculated mean size of the grain was average NPs 105 nm.

EDX spectroscopy is generally used to determine the presence of incorporated specific elements/dopant and their corresponding percentages into the samples. The EDX graph of Fig. 3(b) shows one extra peak in agreement with the chlorine from MnCl₂, corresponding with the calculated and observed crystallographic data.

3.3. Cytotoxic analysis

The cytotoxicity of the prepared Mn:CeO₂ nanocomposites was examined for MCF-7 breast tumorous cell line as shown in Fig. 4. The cell line was incubated with 20 mg/mL solutions of the manufactured Mn:CeO₂ nanocomposites for a day.

The in-vitro analysis on cells of human breast cancer cells and math model simulations have revealed that average NPs 105 nm are translocated proficiently within surficial tissue resulting in harming the structure of cell including strong cellular uptake (Gustafson et al., 2015) with no toxicity is shown in Fig. 5.

3.4. Mathematical analysis

The data was collected using different experiments for different values of ppm NP for each DMEM (Dulbecco’s Modified Eagle’s medium) and Mn-doped CeO₂. The results presented were used for mathematical modeling and curve fitting presented in Fig. 6.

4. Discussion

The XRD patterns, which are coincident with typical values of CeO₂ (JCPDS card no. 43-1002), clearly indicate the formation of a face-centered cubic fluorite structure (Anupriya et al., 2014). The absence of MnO or any other impurity phases reveals the phase purity of the prepared samples and suggests the possible substitution of Mn ions on the Ce sites. Hence, the peak broadening increased due to the decreased particle size, as shown in Fig. 2 (Abbas et al., 2016).

Generally, metal ion doping at optimal level hinders the particle growth. The effect of Mn doping on the size reduction of the CeO₂ nanocomposite can be attributed to grain-boundary pinning affected by dopant ions, which bounds the grain growth by symmetry-breaking effects of the dopant at the boundary, resulting in particles with smaller size. The reduction in NP size after doping has been previously reported (Kumar et al., 2013; Jayakumar et al., 2017). SEM image reveals the cluster formation mechanism with crystal growth. This crystal growth is also confirmed by XRD. The analysis behind the crystal growth in the
The morphology of ceria is discussed by previous reported studies explaining two common mechanisms called Ostwald ripening (OR), and Oriented attachment (OA) (Wu et al., 2008; Lin et al., 2012).

Furthermore, the EDX plot strongly coincident with the typical peak positions of La, Fe, Mn, and O. Ce was also identified by EDX spectroscopy because the energy of Mn (5.88) is very adjacent to that of Ce (6.02) (Abbas et al., 2016).

Synthesized Mn-doped CeO₂ nanocomposite anticancer activity was considerably greater in comparisons to other metal oxide nanostructures. For instance, Alhadlaq et al. reported in 2015 the cytotoxicity and oxidative stress of zinc ferrite in HepG2, A431, and A549 (Alhadlaq et al., 2015); Watanabe et al. discussed in 2013 the inhibition of A549 cancerous cells using Fe₃O₄ magnetic NPs (Watanabe et al., 2013); Alarifi et al. reported in 2014 a 16% inhibition in the % cellular viability of MCF-7 cancerous cells by Fe₂O₃ nanoparticles (Alarifi et al., 2014). In our case, 68% viability was inhibited using 200 ppm of a NP suspension of the Mn:CeO₂ nanocomposite, as shown in Fig. 4.

The highly selective cytotoxic performance of the synthesized Mn:CeO₂ nanocomposite may be of high clinical important because conventional anticancer drugs are often not able to discriminate between healthy and cancerous cells (Abbas et al., 2017). The nanomaterials toxicity is dependent on numerous factors such as composition and type of the nanomaterials, shape, particle size, and surface area, crystallinity, and solubility (Atif et al., 2019; Ahamed et al., 2016; Mehmood Ur Rehman et al., 2017; Mehmood Ur Rehman et al., 2018; Mehmood Ur Rehman et al., 2019). An elevated oxidative stress can be proposed as the main reason for the decrease of cell viability in the present study. Moreover, as can be extracted from Fig. 4, Mn-doped CeO₂ nanocomposites produced considerably higher amount of ROS in the MCF-7 cell line. This enhancement in ROS production has been previously related with the oxygen vacancies. Thus, the presence of structural defects (oxygen vacancies) in the crystal structure of Mn-doped metal oxides nanostructures leads to increased ROS generation (Abbas et al., 2015; Abbas et al., 2015; Iqbal et al., 2019). We previously observed a similar relation for Fe-doped CeO₂; however, no such a connection has been found in the case of Co-doped CeO₂ NPs (Iqbal et al., 2019; Iqbal et al., 2019). Hence, further investigation is needed.

The membrane integrity is disrupted by positively charged NPs and their toxicity is high as compared to negatively charged NPs (Anupriya et al., 2014) due to their low toxicity toward eukaryotic cells. Their easy surface modifications and biocompatibility is shown in Fig. 5. In short, Mn-doped CeO₂ nanocomposites are

**Fig. 3.** (a) Morphological and (b) energy dispersive X-ray spectroscopy analysis of the Mn:CeO₂ nanocomposite.

**Fig. 4.** Cytotoxicity of the Mn:CeO₂ nanocomposite in MCF-7 cancer cell line.

**Fig. 5.** Mechanism of the interaction of nanostructures with cancerous cells.
promising candidates for biosensor applications, drug delivery, and biomedical imaging including bacterial detection and antibacterial activity (Abbas et al., 2016).

The least-square error method was used to model the data for identifying the best possible values of the constants for the linear and exponential models presented in Fig. 6.

The reason for the selection of a biexponential model for DMEM is the nature of the decaying % cell viability with the increase in ppm NP. It can be observed that the fitted model is close to the data points and hence satisfies the 95% confidence level. The values for the constants used in mathematical modeling, which were extracted using Matlab software, are presented in Fig. 6, along with the values of goodness of fit for DMEM. It can be observed that the values of goodness of fit were satisfactory (Fakhar-e-Alam et al., 2020).

Similarly, the data for Mn-doped CeO2 were also modeled using least-square errors; however, unfortunately, lower level of confidence was obtained because only a polynomial function was found to fit the data. Therefore, data were divided into two discontinuous functions (Iqbal et al., 2019) depending on the nature of the data. The first few data points followed a straight line, and a linear model was adopted as the curve-fitting tool. Then, the later portion followed a decaying exponential trend, and hence an exponential model was selected to fit the data. The values of the constants for both functions and a reasonable goodness of fit are included in Fig. 6 (Atif et al., 2020).

5. Conclusions

Mn-doped CeO2 nanocomposites are found to be promising candidates for drug delivery and anticancer therapy. An MnₓCe₁₋ₓO₂ nanocomposite was successfully synthesized by hydrothermal method. Microstructural characterization by XRD, SEM, and EDX verified its morphology as cubic fluorite phase, nanospherical shape, and doping composition. The Mn-doped CeO₂ nanocomposite showed significant cytotoxic activity, demonstrating a 68% growth reduction of MCF-7 cancerous cells upon Mn doping most likely due to generation of oxygen vacancies resulting from ROS production. The experimental results were verified by mathematical modeling generated by applying the least-square error method using Matlab, which yielded a 95% confidence level.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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