Possible bioremediation of arsenic toxicity by isolating indigenous bacteria from the middle Gangetic plain of Bihar, India

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1. Introduction

Arsenic is a toxic metalloid, widely distributed due to natural and anthropogenic activities in the environment. It occurs in four oxidation states (+5, +3, 0, and –3) although the arsenate (As\textsubscript{V}) and arsenite (As\textsubscript{III}) are the most common forms and As\textsubscript{III} is more toxic than As\textsubscript{V} \cite{1,2}. Drinking water and food are the main sources of exposure of arsenic to the consumers, including animals and humans. Arsenic species are deposited in the skin, lungs, kidney, liver, etc. and cause several severe diseases by oxidative stress, altered DNA methylation, altered DNA repair, mitochondrial damage, proliferation of cell, tumour promotion and co-carcinogenesis \cite{1,3}. It is reported that conventional methods such as oxidation or reduction, chemical precipitation, filtration, ion exchange, reverse osmosis and evaporation recovery of cleaning contaminated water are too much expensive and laborious \cite{4}. So, there is need to develop eco-friendly and low cost technique to mitigate the arsenic contamination.

It is well documented that bioremediation is a cost-effective and a comparatively innocuous alternative to physical methods for heavy-metal remediation \cite{3–6}. Bioremediation of arsenic species by microbial community involves their reduction, oxidation, intracellular bioaccumulation and methylation \cite{6}. The arsenite-oxidising ability to the bacteria is provided by aox operon. The expression of the aox operon is controlled by AoxR, after expression of the aox operon AoxAB complex is synthesized and exported to the periplasm. The AoxAB complex, an arsenite oxidase is involved in the oxidation of the As\textsubscript{III} into As\textsubscript{V} \cite{6–8}.

Ganga river, originating from the Himalaya, is one of the natural source of arsenic in the Gangetic plain of Bihar. But there is no proof regarding the natural emission of arsenic in the Ganga plain so far \cite{9}. However, the arsenic is released in the Gangetic plain of Bihar by the natural processes in groundwater from holocene sediments containing clay and silt \cite{10,11}. The concentration of arsenic in Gangetic plain of Bihar found above the permissible limit of 10 ppb \cite{12}. So, there is an urgent need of remediation of these contaminated areas.

As far as our knowledge, study related to bioremediation of arsenic toxicity by employing arsenite-oxidising bacteria from arsenic contaminated groundwater of the middle Gangetic plain, Bihar, India is not available. So considering the importance of work

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in present study, we describe the stimulation of the indigenous bacteria for bioremediation of arsenic toxicity. The bacterial isolates were also evaluated for other heavy metal resistance such as Cr(IV), Ni(II), Co(II), Pb(II), Cu(II), Hg(II), Ag(I) and Cd(II).

2. Material and methods

2.1. Sample collection

The water samples were collected before the rainy season from handpumps of 12 different arsenic contaminated sites of Gangetic plain in the Bihar region. Water samples were collected in two different storage bottles, one sample was treated with 2–3 drops of nitric acid to prevent the metal from precipitation, adsorption and microbial degradation. Another water sample was used for isolation of bacteria hence kept as it is at ice. The selected sites were namely Karja (25°39′09.2″N 84°42′27.7″E), Pararia (25°24′05.9″N 84°34′41.8″E), Bhakura (25°39′21.1″N 84°09′00.2″E), Semaria (25°40′27.8″N 84°43′26.6″E), Keshopur (25°39′54.7″N 84°43′06.6″E), Maner Thana (25°38′51.6″N 84°53′07.6″E), Danapur (25°34′56.4″N 85°02′36.9″E), Badhora (25°22′08.0″N 85°59′58.1″E), Danapur (Auto-stand) (25°38′14.1″N 85°02′42.4″E), Maner Thana (Lodipur) (25°38′52.0″N 84°53′06.8″E), Akauna (25°07′18.5″N 85°24′16.6″E), Science College (25°37′05.6″N 85°10′10.2″E) (Fig. 1). The samples were kept in sterile sample collection plastic bottles and preserved at 4 °C for further use.

2.2. Evaluation of total arsenic in water samples

The collected water samples were analysed for total arsenic present by using MQuant Arsenic Test kit (Merk). Total arsenic concentration was measured semiquantitatively by visual comparison of the reaction zone of the test strip with the fields of a colour scale. The measuring range of the test strip was varied from 5 ppb to 500 ppb.

2.3. Isolation of arsenic resistant bacteria

For isolation of arsenic resistant bacteria, 100 μl of water samples were inoculated in 10 ml of Luria-Bertani (LB) media and incubated at 30 °C for overnight [3,4,18]. The 100 μl of revived culture was spread onto LB-agar plates containing 1.33 mM of As(III). The plates were incubated at 30 °C for 72 h. The colonies, which showed resistance to As(III) and were morphologically different, were picked up and isolated after successful purification by growing repeatedly on LB medium and stored at 4 °C.

2.4. Evaluation of the MIC value

The minimum inhibitory concentration of As(III) for all the 48 isolates was evaluated by growing them on As(III) supplemented medium [18,21]. 2 μl of the freshly revived cultures were spotted onto LB-agar plates supplemented with increasing concentration (1.33–20 mM) of As(III). The plates were then incubated at 30 °C for
72 h and colonial growth was observed, and two isolates (AK1 and AK9) were selected for further work.

2.5. Cellular and morphological characterization

Gram’s staining was done using Gram stains-Kit (Himedia) and cellular structure was observed under a bright field microscope. For morphological study, streaking of the fresh culture was done by using inoculation loop on solidified LB-agar plates and incubated at 30 °C, and colony morphology was recorded after 72 h of incubation [18]. The colony morphology of AK1 and AK9 isolates was observed in terms of their general surface form, elevation, margin, surface texture, colony size, appearance, pigmentation and optical property.

2.6. Biochemical analysis of the isolated strains

Biochemical characterization of AK1 and AK9 isolates was done in terms of production of enzymes such as β-galactosidase, urease, nitrate reductase and carbohydrate utilization. The biochemical tests were done by using Biochemical test kit (Himedia) and analysed by using its result analysis index provided with kit. The experiments were performed in triplicates with freshly revived culture.

2.7. Physiological characterization of the isolated strains

For optimum growth of the bacterial isolates, physiological characterization was carried out on the basis of optimum pH and temperature.

2.8. pH and temperature optimisation

To determine the optimum pH for growth of isolates, 0.4% (v/v) of freshly revived culture was inoculated in 10 ml of LB media adjusted to pH 5–10 and incubated at 30 °C on 180 rpm for 12 h [3,4,18]. After incubation, the optical density was measured at 600 nm by UV–visible spectrophotometer (Evolution 201, ThermoScientific). All the experiments were performed in triplicates.

To optimise the temperature for optimum growth of isolates, 10 ml of LB media was inoculated with 0.4% (v/v) of fresh culture and incubated at 20, 25, 30, 35 and 40 °C. After 12 h of incubation period, absorbance was measured at 600 nm using UV–vis spectrophotometer. The experiments were performed in triplicates.

2.9. Screening of arsenic transforming bacteria by microplate screening assay

Microplate screening assay was used to estimate the efficiency of arsenic transformation by the arsenic-resistant strains [15]. The method is based on the reaction between AgNO₃ and AsⅢ and/or AsⅤ in Tris-HCl which results in colour precipitate ranging from light brownish-red with AsⅤ to light yellow with AsⅢ. The different concentration of AsⅤ/AsⅢ was defined as a colour scale which gives a correlation between the colour intensity and the ratio of AsⅤ and AsⅢ species. To estimate the arsenic transforming efficiency of isolates, AK1 and AK9, to transform the arsenic species, isolates were grown overnight at 30 °C at 180 rpm to reach an optical density of 0.4–0.6. Further, 2 ml of the culture was centrifuged at 2,152g for 15 min. The pellet was washed twice with sterile Milli-Q water. Pellets were further suspended in 1.2 ml of Milli-Q water to prepare a cell suspension. In the microtiter plates, 20 μl cell suspensions and 80 μl of 0.2 M Tris-HCl buffer (pH 7.4) supplemented with AsⅢ/AsⅤ to reach a final concentration 1.33 mM of AsⅢ/AsⅤ. Microtiter plates were further incubated at 30 °C for 24, 48 & 72 h. Adding 100 μl of 0.1 M of AgNO₃ to the wells after the incubation period and left for 15 min for colour precipitate formation and compared with the colour scale.

2.10. Molecular characterization

2.10.1. Genomic DNA isolation

Genomic DNA of isolates was extracted by modified cetyl trimethyl ammonium bromide (CTAB) method [16]. The bacterial strains were grown overnight in 10 ml LB medium at 30 °C with shaking at 180 rpm 2 ml of the culture was centrifuged at 8,609 g for 30 s and then the pellet was suspended in 567 μl of TE buffer (10 mM Tris, 1 mM EDTA), 30 μl of 10% SDS and 3 μl of 20 mg/ml proteinase K and mixed well followed by incubation at 37 °C for 1 h. After incubation, 100 μl of 5 M NaCl and 80 μl of CTAB/NaCl (10% CTAB and 5 M NaCl, 1:1 (v/v)) was gently mixed by inverting the tube and incubated at 65 °C for 10 min and further an equal volume of Phenol:Chloroform:isoamyl alcohol (PCI) (25:24:1, v/v) was added, mixed to emulsify and spin for 5 min at 5510 g. The supernatant was transferred into a fresh tube then PCI treatment was repeated as earlier. The supernatant was incubated with RNase (50 μg/ml) for 30 min at 37 °C followed by PCI treatment as earlier. Further, sodium acetate (0.1 time of the total volume) and isopropanol (0.6 vol of the total volume) was added to precipitate DNA and centrifuged at 12,396 g for 20 min, and supernatant was drained. 70% ethanol was used to clean the DNA followed by centrifugation at 12,396 g for 20 min. Pellet was dried completely and dissolved in 50 μl of TE buffer for storage at −20 °C.

2.10.2. PCR amplification of 16S rDNA

The isolated genomic DNA of the isolates was used as template DNA for PCR amplification of 16S rDNA gene. The 16S rDNA gene fragment was amplified by using 27F (5′-AGAGTTTGATCMTGGCT-CAG-3′) and 1492R (5′-TACGYYTACCTTGTACGACTT-3′) primers. The PCR reaction mixtures contains 1X PCR buffer (G-biosciences), 2.5 mM MgCl₂, 0.2 mM dNTPs, 1 μM each primer, approximately 50 ng DNA templates, 1 unit Tag DNA polymerase (G-biosciences) and molecular grade water to a final volume of 50 μl. PCR condition was; initial denaturation at 95 °C for 5 min, 30 cycles of denaturation (94 °C, 1 min), annealing (56 °C, 1 min), extension (72 °C, 10 min), and the final extension at 72 °C for 7 min and storing was done at 4 °C. All the PCR products were purified on 1.5% agarose gel and further eluted by using QIAQUICK Gel Extraction Kit for DNA sequencing.

2.10.3. DNA sequencing and phylogenetical analysis

The ~1.5 kb PCR amplified products 16S rDNA was used for DNA sequencing (Eurofins Genomics India Pvt Ltd., Bangalore, India) with the same primers: 27F and 1492R. The NCBI nucleotide BLAST (http://blast.ncbi.nlm.nih.gov/Blast.cgi) was used to search similar nucleotide sequences of 16S rDNA gene. The 16S rDNA genes of the isolates and similar sequences retrieved from the NCBI database were aligned using CLUSTAL W and the phylogenetical analysis was

### Table 1

| Primer    | Primer sequence                  | Tₙ (°C) |
|-----------|----------------------------------|--------|
| aoxRF     | 5′-AATCCGCTATCCAGGCATTTCGC-3′    | 59     |
| aoxRR     | 5′-TGGCCTCTGCGAACCTACTAGA-3′     | 59     |
| aoxAF     | 5′-TCCGCTACCCCTACCTACCC-3′       | 53     |
| aoxF      | 5′-TCCGCCGCGCGGACGAGTF-3′        | 53     |
| aoxC      | 5′-TGCGCATCGGCCGACGAGTF-3′       | 53     |

a. Annealing temperature used in this study.
made using the neighbour-joining method. The sequence analysis was carried out by using the software MEGA 6.

2.10.4. PCR amplification of aox genes

The primers used for PCR amplification of aox genes are listed in Table 1 [16]. The PCR reaction mixtures were composed of 1x PCR buffer (G-biosciences), 2.5 mM MgCl2, 0.2 mM dNTPs, 1 μM each primer, approximately 50 ng of DNA templates, 1 units Taq DNA polymerase (G-biosciences) and molecular grade water to a final volume of 50 μl. The PCR conditions consisted of 5 min of initial denaturation at 95 °C, 30 cycles of 1 min at denaturation at 94 °C, 1 min annealing as mentioned in Table 1, 1.5 min extension at 72 °C, and the final extension at 72 °C for 7 min. The PCR was conducted in a ProFlex PCR system (Life Technologies, USA) and the PCR product was identified by gel electrophoresis using 1.5% agarose gels.

2.10.5. Effect of AsIII on the growth of isolates

The effect of AsIII on the growth of isolates was determined in the presence of AsIII and without AsIII act as control. 0.4% (v/v) of the fresh culture was added to 10 ml of LB media supplemented with 1.33 mM of AsIII and incubated at 30 °C on 180 rpm [3,4]. The optical density was measured at 600 nm using UV–vis spectrophotometer at a regular interval of 1 h. The experiments were performed in triplicates.

2.10.6. Heavy metal tests

The ability of isolates, AK1 and AK9, to tolerate the different heavy metal salts, like Ag (as AgNO3), Hg(II) (as HgCl2), Co(II) (as CoCl2⋅6H2O), Pb(II) (as Pb(NO3)2), Cd(II) (as Cd(NO3)2), Cu(II) (as CuSO4⋅5H2O), Ni(II) (as Ni(NO3)2⋅6H2O) and Cr(IV) (as K2Cr2O7), was evaluated at different concentrations [3,4]. 10 μl of the fresh culture was spread over LB-agar plates supplemented with increasing concentration from 0.5–20 mM of heavy metal salts. The plates were incubated at 30 °C for a period of 72 h. Further, the colony growth was observed and MIC of the heavy metals was recorded. All the experiments were performed in triplicates.

3. Results

3.1. Arsenic analysis of the water sample

Water samples were collected from the middle Gangetic plain of Bihar and analysed for the presence of arsenic. The results show that total arsenic in the water sample varied from 0 to 100 ppb (Table 2).

| Sl. No. | Sample         | Total arsenic (ppb) |
|--------|---------------|---------------------|
| 01.    | Karra         | 100                 |
| 02.    | Parasia       | 100                 |
| 03.    | Bhakhura      | 0                   |
| 04.    | Semaria       | 25                  |
| 05.    | Keshopur      | 50                  |
| 06.    | Maner Thana   | 5                   |
| 07.    | Danapur       | 5                   |
| 08.    | Badhora       | 5                   |
| 09.    | Danapur (auto stand) | 0     |
| 10.    | Maner Thana (lodipur) | 5     |
| 11.    | Akoura        | 25                  |
| 12.    | Science College | 10                 |

3.2. Isolation of arsenic resistant bacterial strains

A total of 48 morphologically distinct arsenic transforming bacterial strains were isolated. They have potential to tolerate 1.33 mM of AsIII stress. It was observed that the MIC value of AsIII for the isolated strains varied from 3 to 15 mM. Out of 48 isolates, two isolates namely AK1 and AK9 showed highest tolerance value for the AsIII and having different colony morphology. The observed MIC value of AsIII for AK1 and AK9 was 13 mM and 15 mM respectively. So, for further study AK1 and AK9 were selected.

3.3. Cellular and colony morphology

The cellular morphology of isolates was observed under a bright field microscope. AK1 was a gram negative coccus in nature, whereas AK9 was a gram negative bacillus. The colony morphology of AK1 was flat, filamentous margin, rugose surface texture with white Colour shiny appearance and AK9 was convex, entire margin, smooth surface texture with a tan colour shiny appearance.

3.4. Biochemical analysis of isolated strains

Biochemical characterization of isolates, AK1 and AK9, was done in terms of production of enzymes; e.g. β-galactosidase, urease, nitrate reductase, and carbohydrate utilization (Table 3).

3.5. Physiological characterization of isolated strains

3.5.1. pH and temperature optimization for growth

The strains, AK1 and AK9 were grow at pH range of 6 to 9 but the optimal growth was observed at pH 7.0 (Fig. 2). The AK1 strain was growing at 20, 25 and 30 °C but the highest growth was observed at 30 °C. AK1 was unable to grow at 35 and 40 °C. Whereas AK9 can grow on 20, 25, 30, 35 and 40 °C but the optimum temperature for it was 30 °C (Fig. 3).

3.5.2. Screening for arsenic transforming bacteria

The ability of bacterial isolates to oxidise AsIII or reduce AsV was observed by AgNO3 test based microplate method. The interaction of AgNO3 with Tris-HCl results in precipitate formation.

| Sl. No. | Tests       | AK1 | AK9 |
|--------|-------------|-----|-----|
| 01.    | Indole      | –   | –   |
| 02.    | Voges- Proskauer | –  | –   |
| 03.    | Citrate utilization | –  | –   |
| 04.    | Lysine utilization | –  | –   |
| 05.    | Ornithine utilization | –  | –   |
| 06.    | Arginine utilization | –  | –   |
| 07.    | Nitrate reduction | +  | +   |
| 08.    | Malonate utilization | +  | +   |
| 09.    | Urease      | +   | +   |
| 10.    | Phenylalanine deamination | –  | –   |
| 11.    | H2S production | –  | –   |
| 12.    | ONPG        | –   | –   |
| 13.    | Glucose utilization | +  | +   |
| 14.    | Mannitol utilization | +  | +   |
| 15.    | Xylose utilization | –  | –   |
| 16.    | Inositol utilization | –  | –   |
| 17.    | Sorbitol utilization | –  | –   |
| 18.    | Rhamnose utilization | –  | –   |
| 19.    | Sucrose utilization | +  | +   |
| 20.    | Lactose utilization | –  | –   |
| 21.    | Arabinose utilization | –  | –   |
| 22.    | Adonitol utilization | –  | –   |
| 23.    | Raffinose utilization | –  | –   |
| 24.    | Salicin utilization | –  | –   |

Table 2

Total arsenic concentration in water samples.

Table 3

Biocatalytic characterization of the strains AK1 and AK9.
The precipitates containing arsenic were coloured from light yellow colour in the presence of AsIII to light brown-red in the presence of AsV. According to AgNO₃ test, both AK1 and AK9 strains, have potential to transform arsenic from AsIII to AsV and AsV to AsIII. The colour scale which, can be easily distinguished by eye, was developed to show different ratios of AsV/AsIII in the solution can be easily distinguished by naked eye (Fig. 4(a)). The arsenic transforming ability of both the strains was observed after 24, 48 and 72 h of incubation at 30 °C (Fig. 4(b)). Both AK1 and AK9 reduced 1.33 mM of AsV to AsIII completely and oxidised the 1.33 mM AsIII to AsV upto 25% after the incubation of 72 h.

3.6.4. PCR amplification of aox genes

The presence of the aox genes were evaluated using PCR by specific primers from the genomes of the bacterial isolates. Both the strains, AK1 and AK9, showed the presence of the aox operon genes. The PCR amplified products show aoxR (two bands, 750 bp and 670 bp), aoxB (~1600 bp) and aoxC (~550 bp) (Fig. 7). The agarose gel image was processed in Bio-Rad’s ChemiDoc™ MP and analysed with ImageLab Version 4.1.

3.6.5. Effect of AsIII on bacterial growth

Both isolates follow a sigmoid pattern of growth (Fig. 8). It was observed that strain AK1 having growth rate constant (k) calculated in the absence of AsIII was 0.641 h⁻¹ (a doubling time of 1.56 h), and in the presence of AsIII it was 0.495 h⁻¹ (a doubling time of 2.02). Strain AK9 having growth rate constant (k) calculated in the absence of AsIII was 0.850 h⁻¹ (a doubling time of 1.176 h), and in the presence of AsIII it was 0.691 h⁻¹ (a doubling time of 1.447). The results showed that in the presence of AsIII, 23% and 19% reduction in cellular growth of AK1 and AK9 respectively. The doubling time of both isolates, increases in the presence of AsIII.

3.6.6. Evaluation of metal tolerance

The ability of isolates, AK1 and AK9, to tolerate other toxic metal salts, like Ag(I), Hg(II), Pb(II), Ni(II), Cu(II), Cd(II), Co(II) and Cr(IV) was observed at different concentrations on LB-agar medium. AK1 and AK9, both have potential to tolerate other heavy metals and the MIC value of heavy metals was recorded (Table 4). The ability of isolates to tolerate the different heavy metal gives an opportunity to use them in metal contaminated water, soil and effluents for its bioremediation.
4. Discussion

Microbial transformation, i.e. oxidation and reduction of toxic AsIII and AsV has evident potential in the treatment process of arsenic from contaminated water. It is emerging as superior to conventional chemical and physical methods. River originating from the Himalaya, like the Ganga river, is the natural source of release of arsenic into the environment [7]. Thus, microorganisms present in this aquatic system have developed some arsenic detoxification mechanisms for their survival. It is well documented that the bacteria have potential for survival in metal stress condition [17,18]. This suggests a wide distribution of arsenic transforming bacteria may be present in Ganga river that has the potential for arsenic detoxification.

In the present study, two prominent arsenic oxidising bacterial strains AK1 and AK9 belong to genus Pseudomonas have been reported and isolated from middle Gangetic plain and both isolates AK1 and AK9 were able to transform AsV to AsIII and AsIII to AsV.

Many arsenic transforming bacteria were widely reported from different aquifers throughout the world [16,19,20], but up to our knowledge no such studies have been done earlier from arsenic prominent middle Gangetic Plain of Bihar.
Isolates AK1 and AK9 in this study were resistant of AsIII up to 13 mM and 15 mM respectively. Both show complete aerobic reduction of AsV after 48 h of incubation. Bacteria belongs to genus *Pseudomonas* can reduce AsV aerobically up to 200 mM [21]. AK1 and AK9 were oxidised 25% of AsIII after 72 h of incubation. Some genus including *Pseudomonas* can oxidise 1 mM of AsIII within 25–30 h [21]. This property permits these bacteria to survive with the elevated arsenic concentrations in native groundwater.

The detection of *aox* genes in the two bacterial isolates, AK1 and AK9, show their potential to oxidise AsIII to AsV. The genes, *aoxR*, *aoxB* and *aoxC* are involved arsenite oxidation [6,8]. Chang et al. also showed the *aoxR* and *aoxB* genes amplified from the genomes of Damyang and Woopo wetlands bacteria [16]. Further growth and physiology of isolated strains were carried out. pH and temperature are the major environmental factors that have a great importance in the growth and metal accrual properties of the bacterial strains [18]. It was observed that both the arsenic resistant isolates, AK1 and AK9, show optimum growth at pH 7.0 and temperature 30 °C. Similarly, it was reported that arsenic oxidising bacteria can grow at 40 °C and 50 °C [19]. The doubling time of the AK1 decreases while it increases in case of AK9 in the presence of AsIII which indicates that AK9 was more resistant to AsIII than AK1. Further, the growth rate of both, AK1 and AK9, decreases in the presence of AsIII. Subsequently, the doubling time of several bacterial strains increases in the presence of arsenic and growth rate decreases [22]. In addition to arsenic resistance, both isolate can grow in presence of other toxic heavy metals. Both, AK1 and AK9, were also resistance to Pb(II), Cu(II), Ni(II) and Cr(IV) with MIC in the range of 4–8 mM (Table 4). Both isolates were showing less tolerance for Ag, Hg(II), Cd(II) and Co(II) up to the concentration 0.5 mM. The heavy metal tolerance properties of these two isolates make them to be a good candidate for bioremediation of toxic heavy metals.

The present work described the first evidence of arsenic transformation from genus *Pseudomonas* from the middle Gangaetic plain of Bihar, India. The strains were able to oxidise AsIII and reduce AsV at high rate. Thus, these new strains could be an excellent candidates for application in arsenic remediation processes. This study provides the first look for the relevant microorganism that exists in the middle Gangaetic plain of Bihar, India. Our observations showed that isolates, AK1 and AK9, capable in biotransformation of arsenic were naturally present in the water sample. The identified strains, seem to be very much familiar to arsenic concentrations present in the water samples.
information is important since, the effectiveness of in situ bioremediation technology depends on the activities of the native microbial communities present in polluted sites.

Conflict of interest

No

Acknowledgements

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Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at https://doi.org/10.1016/j.btre.2018.02.002.

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