Research Article

GC-MS Profiling, In Vitro Antioxidant, Antimicrobial, and In Silico NADPH Oxidase Inhibition Studies of Essential Oil of Juniperus thurifera Bark

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Juniperus thurifera is a native species to the mountains of the western Mediterranean region. It is used in traditional medicine as a natural treatment against infections. The present study aimed to carry out the chemical analysis and evaluate the antioxidant, antimicrobial, and in silico inhibition studies of the essential oils from Juniperus thurifera bark (EOEJT). Chemical characterization of EOEJT was done by gas chromatography (GC-MS). We have performed three antioxidant assays (Reducing power (FRAP), 2, 2-diphenylpicrylhydrazyl (DPPH), and total antioxidant capacity (TAC)) of the EOEJT. We next evaluated the antimicrobial activity against the NADPH oxidase, which was carried out to help evaluate the inhibitory effect of EOEJT against NADPH oxidase. Results of the GC/MS analysis revealed seven major compounds in EOEJT wherein muurolol (36%) and elemol (26%) were the major components. Moreover, EOEJT possessed interesting antioxidant potential with an IC50 respectively of 21.25 ± 1.02 μg/mL, 481.02 ± 5.25 μg/mL, and 271 μg EAA/mg in DPPH, FRAP, and total antioxidant capacity systems. Molecular docking of EOEJT in NADPH oxidase active site showed inhibitory activity of α-cadinol and muurolol with a glide score of −6.041 and −5.956 Kcal/mol, respectively. As regards the antibacterial and antifungal capacities, EOEJT was active against all tested bacteria and all fungi, notably, against Escherichia coli K12 with an inhibition diameter of 21 mm and a MIC value of 0.67 mg/mL, as well as against Proteus mirabilis ATCC 29906 with an inhibition diameter of 18.33 ± 1.15 mm and a MIC value of 1.34 mg/mL. A more pronounced effect was recorded for the fungal pathogens Fusarium oxysporum MTCC 9991 with inhibition of 37.44 ± 0.28% and MIC value of 6.45 mg/mL, as well as against Candida albicans ATCC 10231 with an inhibition diameter of 20.33 ± 1.15 mm.
1. Introduction

Morocco is the most biodiverse country in the western Mediterranean region [1]. Within it, the eastern half of the forest formations of the Middle Atlas is a veritable mosaic, with a wide range of forest species including cedar, holm oak, Phoenician juniper as well as Juniperus thurifera [2]. The Bouiblane site made up of two distinct regions (Bouiblane I and Bouiblane II) is part of the Sites of Biological and Ecological Interest (SBEI) of the Moroccan Eastern Middle Atlas. This locality stands out by its different agro-climatic conditions, its floristic diversity, and its significant but unquantified rate of endemism, as well as its wide range of plant forms [3]. Juniperus thurifera is a variably sized tree that belongs to the family Cupressaceae. This species is native to mountainous areas of the Mediterranean basin’s western region [4]. In Morocco, Juniperus thurifera grows more particularly in the High and Middle-Atlas Mountains [5, 6]. The two arboreal plants that can grow with it are Quercus ilex and Cedrus atlantica. Such hardy species may tolerate a wide range of weather conditions and may reach up to 19 meters tall and live for up to 800 years. The thurifer covers an estimated area of almost 20,000 hectares. It is the most regressed forest species in Morocco, with a 90 percent regression from its potential range [7].

Different phytoconstituents have been isolated from this species, and include mainly polyphenols, essential oils, and sterols [8, 9]. Such metabolites exhibited potential bioactivities such as anti-inflammatory, and antibacterial agents among other wide therapeutic applications [8–10].

Multidrug-resistant organisms have become a major concern and the situation is far from being mastered. Therefore, the research for antibacterial from natural origins has become a vivid area of exploration. Essential oils are combinations of volatile secondary metabolites that occur naturally in plants, which mostly represent the various smells and scents of a variety of plants [10]. A myriad of published reports has provided evidence of their potent bioactivities as antioxidants, antibacterials, antifungals, and insecticides [8–12].

This study aimed at characterizing the bark essential oils of Juniperus thurifera by GC/MS. Next, we conducted the antioxidant potential of essential oils in vitro and explore in silico as NADPH oxidase inhibitors. Finally, the antibacterial and antifungal effects against drug-resistant microbial strains were also described.

2. Materials and Methods

2.1. Plant Material. Juniperus thurifera (Figure 1) was collected from a mountainous area in the Middle Atlas, Morocco before being identified by a botanist and given the voucher number (FFT/02D20). Next, the bark was isolated from the plant branches, and then cleaned and dried by use of an oven at 35°C before grinding into powder by use of an electric grinder [13].

2.2. Chemicals. Ferric chloride (FeCl₃), sodium phosphate, ammonium molybdate, butylated hydroxytoluene (BHT), 2, 2-diphenylpicrylhydrazyl (DPPH), and 2, 3, 5-triphenyltetrazolium chloride (TTC). Malt extract (ME), sodium chloride (NaCl), trichloroacetic acid (TCA), potassium ferricyanide (K₃Fe(CN)₆), agar, erythromycin, fluconazole, and dimethylsulfoxide (DMSO) were bought from Sigma-Aldrich (St. Louis, MO, USA).

2.3. Extraction of EOEJT. Using a Clevenger device, 200 g of the plant powder was immersed in 800 mL of water before being mixed and heated at boiling temperature for two hours. Next, the essential oils (EOEJT) were recovered and placed at 4°C until further use [13, 14].

2.4. Chromatographic Analysis of EOEJT by GC/MS. Analysis was done using a gas chromatograph (GCMS-TQ8040 NX Shimadzu brand) with an apolar capillary column (RTxi-5 Sil MS-30.00 m × 0.250 mm ID × 0.250 μm Film thickness). One microliter of EOEJT was used for chromatographic analysis after being diluted in hexane (10:100 dilution). The oven temperature program was initially set to 40°C for 2 minutes and followed by 260°C for 10 minutes, then increased by 5°C/min up to 280°C. Finally, the temperature was held at 280°C for 10 minutes. The carrier gas was nitrogen and the flow rate was 1 mL/min. The temperatures of the injector and detector were adjusted to 250 and 280°C. The ionization energy was set to 70 eV, the ion source temperature was set at 200°C, the interface line temperature was 280°C, and the scan mass range was m/z 40–650.00. Volatile compounds of essential oil were identified by the use of the Kovats index and NIST-MS Search Version 2.0 [13, 15].

2.5. Antioxidant Activity of EOEJT. The antioxidant capacity of EOEJT was assayed by the use of three systems DPPH, TAC test, and FRAP, respectively. EOEJT were prepared in a dilution series from 1 μg/mL to 1000 μg/mL in methanol; thus, positive controls of quercetin and BHT were used in this evaluation and were prepared under the same conditions as the EOEJT.

2.5.1. DPPH Test. Free radical powder was prepared in methanol [16,17]. Briefly, 0.004 mg in 100 mL. 100 μL of the sample (EOEJT, BHT, and quercetin) were diluted in methanol before being added to 750 μL DPPH, while a solution consisting of 750 μL DPPH and 100 μL methanol was used as a negative control. The reaction media were
incubated for 25 minutes, and the optical density (DO) was measured at $\lambda = 517$ nm. The antioxidant efficiency was determined according to the following formula:

$$EA(\%) = \left(1 - \frac{D_s}{C_t}\right) \times 100.$$  

(1)

EA: antioxidant efficiency in percent (%); $D_s$: optical density of sample; $C_t$: optical density of negative control (White).

2.5.2. FRAP Test. This test was assayed by preparing a reaction medium using phosphate buffer with 200 mM-pH $= 6.6$ and $K_3Fe(CN)_6$ (1%) [14,18]. Briefly, 200 $\mu$L of each solution was mixed with 50 $\mu$L EOEJT before being incubated at 50 °C for 30 minutes. To prepare for analysis, the sample was remixed with 200 $\mu$L of trichloroacetic acid (10 mg/mL), 500 $\mu$L of distilled water (H$_2$O), and 100 $\mu$L of FeCl$_3$ (1 mg/mL). At 700 nm, absorbance was measured against a negative control containing 50 $\mu$L of methanol without EOEJT. The EC-50 effective concentration was used to present the results.

2.5.3. TAC Test by Use of Phosphomolybdenum Method. Briefly, EOEJT was combined with 1 mL of a reagent solution before being heated at 95°C for one hour and then cooled at room temperature for 20 minutes. The optical density was measured at $\lambda = 695$ nm against a negative control containing 25 $\mu$L of methanol without EOEJT, while ascorbic acid was used as a positive reference. Results of the TAC assay were expressed in micromilligrams per milligram of the sample ($\mu$g AAE/mg) [19].

2.6. Antimicrobial Capacity

2.6.1. Method for Assessing Antimicrobial Capacity. The disk diffusion assay was used to determine whether or not EOEJT has antimicrobial capabilities [20]. In a nutshell, the double-layer approach was used in order to inoculate bacterial and fungal strains onto Petri plates that contained Mueller–Hinton (MH) agar and malt extract (ME), respectively. The inoculum was prepared from fresh cultures grown in MH and ME media with (0.9% NaCl). Next, Wattman paper discs measuring sex millimeters in diameter were immersed in 20 $\mu$L of EOEJT before being placed on Petri dishes surfaces that were previously inoculated with bacteria (10$^6$ to 10$^8$ CFU/mL) and fungi. Erythromycin and fluconazole served as positive controls, respectively, for bacteria and fungi. Subsequently, Petri dishes inoculated with bacteria and fungi were meticulously incubated at 37°C and 30°C for bacterial and fungi strains, respectively [21]. After 24 hours and 48 hours postinoculation, respectively, inhibition diameters and inhibition percentages were computed for bacterial strains and $C. albicans$. However, after 7 days postinoculation, these calculations were performed for $A. flavus$, $F. oxysporum$, and $A. niger$ [13].

2.6.2. Minimum Inhibitory Concentration (MIC). The MIC of EOEJT against microbial strains was determined using the microdilution method [22]. Briefly, the microplates were prepared in sterile conditions, each sterile 96-well microplate was labeled, and then a 100 $\mu$L of EOEJT with a 1:10 (V/V) dilution of DMSO (10 %) was pipetted into the first column of the plate to perform a series of dilution using sterile MH for bacteria and sterile ME for fungi, and lastly, 30 $\mu$L of microbial suspension of each strain (108 CFU/mL) was added to wells. After an incubation period ranging from 48 hours to 7 days for fungi and 24 hours for bacteria at 37°C [13, 22], the MIC was ascertained by using the colorimetric method (TTC 0.2%), [22, 23].

2.7. Molecular Docking

2.7.1. Ligand Preparations. For ligand preparations, we used the PubChem database to download all the molecules identified by GC/MS in EOEJT in SDF format. Next, the OPLS3 force field was applied, and the LigPrep tool in Schrödinger’s Maestro 11.5 software was used to prepare the structures. Based on the ionization states at pH 7.0-2.0, a total of 32 stereoisomers were generated for each ligand.

2.7.2. Protein Preparation. For receptor preparation, the three-dimensional crystal structure of NADPH oxidase was downloaded from the PDB data bank in PDB format. Protein Preparation Wizard of Schrödinger-Maestro v11.5 was used to add hydrogen atoms to heavy atoms, transform selenomethionine into methionine, and remove water. Next, the OPLS3 force was used for minimization, and fixing the maximum heavy atoms RMSD (root mean square deviation) at 0.30 Å [24].

2.7.3. Receptor Grid Generation. The receptor grid was set at the following coordinates: $X = 17.584$, $Y = 9.05$, and $Z = 51.541$. When the volumetric spacing performed was
20 × 20 × 20, the glide of Schrödinger-Maestro v11.5 was used for SP flexible ligand docking [24].

2.7.4. Receptor Grid Generation. The receptor grid was set at the following coordinates: X = 17.584, Y = 9.05, and Z = 51.541. When the volumetric spacing performed was 20 × 20 × 20, the glide of Schrödinger-Maestro v11.5 was used for SP flexible ligand docking. The ligand was coupled to the grid box produced from protein using "Standard precision" (SP), while SP GScore was used to evaluate the results [24].

2.7.5. Glide Standard Precision (SP) Ligand Docking. In ligand docking, the penalties were applied to noncis/trans amide bonds. The partial charge cutoff was set to 0.15 and the Van der Waals scaling factor and partial charge cutoff were set to 0.80 for each ligand atom. The energy-minimized poses presented by the glide score were used to calculate the final score. The best-docked pose with the lowest glide score value was recorded [25].

3. Statistical Analysis

Data were all provided as the mean value and standard deviation of tests performed in triplicate. GraphPad Prism was used to perform statistical analysis. Levine’s test was used to verify homogeneity, while Shapiro–Wilks test was used to verify normality. Analysis of variance (ANOVA) flowed by Tukey’s HSD test was used to arrange multiple comparisons. p values lower than 0.05 were regarded to indicate a statistically significant difference.

4. Results

4.1. Phytochemical Identification of EOEJT’s by GC/MS. The yield of EOEJT’s was about 0.89%, and it is somewhat comparable to what was previously found in the leaves (0.96%) of Juniperus thurifera [13]. Seven compounds were identified with a dominance of eudesmane-type crypto-meridiol (37.02%) followed by a cadinane-type muurolol (36.31%) and elemene-type sesquiterpenoid which is elemol (26.93%). The representative GC-MS total ion chromatography (TIC) of EOEJT from the bark of Juniperus thurifera is shown in Figure 2 and Table 1.

4.2. Antioxidant Activity of EOEJT. By use of the DPPH assay, EOEJT showed remarkable antioxidant potency in a dose-dependent manner. The concentration of 50 μg/mL scored an inhibition of about 75%, 85%, and 70%, respectively, for EOEJT, BHT, and quercetin. At a higher concentration, 250 μg/mL, the inhibition percentage was found to be 89.21% (EOEJT), 94% (BHT), and 92% (quercetin), Figure 3(a) Depicts the effectiveness of the tested products, which were determined by the MIC of 50% of free radicals (IC50). The calculated IC50 were 21.25 ± 1.02 μg/mL, 17.25 ± 1.20 μg/mL, and 20.15 ± 1.30 μg/mL, respectively, for EOEJT, BHT, and quercetin (Figure 3(b)).

Concerning the FRAP method, EOEJT showed also significant antioxidant effects in a doses-dependant manner compared to the positive controls BHT and quercetin. In this respect, 50 μg/mL of EOEJT or BHT, or quercetin revealed an optical density of 0.33, 0.39, and 0.36, respectively (Figure 3(c)). The EC50 of EOEJT, BHT, and quercetin were 481.02 ± 5.25 μg/mL, 214.08 ± 2.51 μg/mL, and 189.11 ± 2.20 μg/mL, respectively (Figure 3(d)). The antioxidant power of EOEJT was slightly lower than that recorded for J. thurifera leaves in previous work with an EC50 of 190 μg/mL [13].

Total antioxidant power was determined using the ammonium phosphomolybdate (TAC) method. Results showed that EOEJT had a good antioxidant capacity since it scored 271 μg AAE/mg, while PHT used as positive control scored 263 μg EAA/mg (Figure 3(e)). These results were comparable to EOs from Dittrichia viscosa, which revealed 192 μg AAE/mg as TAC [26]. In addition, the EOs of the leaves of Lavandula dentata scored 81.28 μg AAE/mg [27]. EOs from Withania frutescens also scored 91 μg AAE/mg [14].

4.3. Antibacterial Capacity. EOEJT showed promising antibacterial results when compared to erythromycin, particularly vs. E. coli wherein we noted an inhibition zone diameter of 21 ± 0.00 mm and MIC of 0.67 ± 0.00 mg/mL, and against P. mirabilis with an inhibition diameter of 18.33 ± 1.15 mm and a MIC of 1.34 ± 0.00 mg/mL (Figure 4, Figure 5 and Table 2). The antibacterial activity of EOEJT may be due to their physicochemical compositions, most importantly the presence of bioactive molecules such as muurolol, elemol, and pinene.

4.4. Antifungal Capacity. For antifungal susceptibility testing, in vitro evaluation of EOEJT against harmful fungal strains was conducted by the use of the diffusion assay. We found a potent antifungal potency with an inhibition percentage of 37.44 ± 0.28% and MIC value of 6.45 ± 0.00 mg/mL against F. oxysporum compared to fluconazole (Figure 6 and Table 3). Similarly, EOEJT exhibited significant activity against C. albicans with an inhibition zone diameter of 20.33 ± 1.15 mm and a MIC value of 0.67 ± 0.00 mg/mL (Table 3). Besides, EOEJT showed a fungistatic activity vs. A. niger and F. oxysporum and a fungicidal activity vs. C. albicans.

The antimicrobial effect of EOEJT was significant in comparison with one bacterial (erythromycin) and one fungal (fluconazole) specific antimicrobials. The results showed that the EOEJT test was more effective against Gram-negative or Gram-positive bacterial strains. Moreover, the principal component analysis (Figure 7) indicated that all bacterial and fungal strains used for testing showed almost similar sensitivities to EOEJT except for A. flavus, which was resistant to both EOEJT and fluconazole.

4.5. Molecular Docking. Molecular docking was accomplished to understand the interaction profile between EOEJT and NADPH oxidase. Among all molecules studied,
Figure 2: Gas chromatographic profile of EOEJ by GC/MS.

Table 1: Tabulation of the GC-MS analysis of EOEJ.

| Peaks | RT  | Name          | Area (%) | Calculated RI | Literature RI | Molecular structure |
|-------|-----|---------------|----------|----------------|---------------|---------------------|
| 1     | 8.36| α-Pinene      | 14.76    | 948            | 939           | ![Molecule 1](image1) |
| 2     | 29.26| Elemol       | 26.93    | 1528           | 1529          | ![Molecule 2](image2) |
| 3     | 31.72| γ-Eudesmol  | 5.23     | 1664           | 1662          | ![Molecule 3](image3) |
| 4     | 32.05| α-Cadinol    | 4.42     | 1650           | 1654          | ![Molecule 4](image4) |
| 5     | 32.37| Muurolol     | 36.31    | 1640           | 1642          | ![Molecule 5](image5) |
| 6     | 33.61| Eicosane     | 4.55     | 2007           | 2009          | ![Molecule 6](image6) |
Table 1: Continued.

| Peaks | RT  | Name         | Area (%) | Calculated RI | Literature RI | Molecular structure |
|-------|-----|--------------|----------|---------------|---------------|---------------------|
| 7     | 37.02 | Cryptomeridiol | 7.78 | 1811 | 1813 | ![Molecular Structure](image) |

Total 99.98%

**Figure 3: Continued.**
Figure 3: (a, b) Antioxidant capacity using DPPH, (c, d) FRAP method, (e) and total antioxidant capacity.

Figure 4: The antibacterial capacity for EOET on solid media (disc diffusion method).

Figure 5: The antibacterial capacity of EOET in a solid medium (disc diffusion method), (means ± SD, n = 3) marked with the same letter for each strain indicated no significant difference at \( p \leq 0.05 \).
α-cadinol, muurolol, gamma-eudesmol, elemol, and α-pinene present the highest Glide score value of −6.041, −5.956, −5.542, −4.538, and −4.358 kcal/mol, respectively (Table 4).

Regarding the nature of the bonds between EOJT and the active site of NADPH oxidase, the molecules which present lower binding energy include α-cadinol, γ-eudesmol, and muurolol, each one, establishing two hydrogen bonds with the ASP 282 and LYS 134 residues and elemol which formed a single hydrogen bond with residue ASP 282 (Figures 8 and 9).

5. Discussion

In the present work, EOs extracted from Juniperus thurifera were evaluated for their chemical composition, antioxidant, and antimicrobial effects as no previous work dealing with these pharmacological activities of EOs extracted from Juniperus thurifera grows in Morocco to the best of our knowledge. Regarding EOJT, Juniperus thurifera leaves were more qualitatively richer in terpenic compounds and identified 31 compounds with a dominance of α-thujene (25%), elemol (12%), and muurolol (12%) [13]. Previously, 24 compounds were reported to account for 99.46% of the mass of Juniperus thurifera EOs [28]. Such differences may be due to climatic factors variation (altitude latitude, substrate, etc.), harvest period, organs explored (leaves, stem, and bark), as well as the extraction method used [29]. The assessment of oxidative stress is becoming increasingly significant since this particular type of oxidation has been connected to a wide range of health issues, including rheumatoid arthritis, atherosclerosis, diabetes, cancer, and aging [30–32]. It has been reported that the hydroxyl function present in the phytochemical compositions of EOs is responsible for their antioxidant capacities.

Terpenes and phenolic components in EOs are strong antioxidant agents [33–38], which is in agreement with our study. The phytochemical diversity in EOTJ (Table 1), may be responsible for the antioxidant efficiency, whether major or minor compounds may work in synergistic ways [16]. Recent studies have shown the richness of Juniperus thurifera of terpinene, which is considered among the compounds that have been shown to increase the antioxidant power [39].

Concerning the antibacterial power, taking a closer look at the published literature supported that muurolol exerted some high biological activity [41]. Elemol compound, on the other side, has also shown significant antibacterial activity against pathogenic bacteria [40]. EOEJT showed that the essential oils exhibited significant antibacterial capacity against P. aeruginosa and other multidrug-resistant bacteria [42,43]. More particularly, it also had strong antibacterial activity against S. aureus and E. coli strains [44]. The current results were in agreement with the results reported by Rahhal et al. [45], who showed that Juniperus thurifera essential oils exhibited substantial antibacterial capacity against bacteria strains, particularly against S. aureus (31.12 ± 3.11 mm of inhibition), E. coli (13.23 ± 2.59 mm of inhibition), and P. aeruginosa (18.27 ± 2.29 mm of inhibition). Our results are also in agreement with several other studies such as the study by Zeraib et al. [46], which reported that S. aureus was highly sensitive to the EO of Algerian Juniperus thurifera, and also the study by Bahri et al. [47] demonstrated that the essential oil of Juniperus thurifera had potent antibacterial activity against S. aureus ATCC 33862 (inhibition zone diameter: 27 mm; minimum inhibitory concentration: 450 μg/mL), E. coli ATCC 25922 (inhibition zone diameter: 25.6 mm; minimum inhibitory concentration: 530 μL/mL), and P. mirabilis ATCC 7002 (inhibition zone diameter: 18.8 mm).

**Table 2:** antibacterial capacity of EOJT in a liquid medium (microdilution method).

|                          | Staphylococcus aureus ATCC 6633 (mg/mL) | Escherichia coli K12 (mg/mL) | Bacillus subtilis DSM 6333 (mg/mL) | Proteus mirabilis ATCC 29906 (mg/mL) |
|--------------------------|----------------------------------------|-----------------------------|----------------------------------|-------------------------------------|
| **EOJT**                 | 1.34 ± 0.00a                           | 0.67 ± 0.00a                | 2.69 ± 0.00a                     | 1.34 ± 0.00a                       |
| Erythromycin             | b                                      | b                           | b                                | b                                   |
| DMSO 10%                 | c                                      | c                           | c                                | c                                   |

(Mean ± SD, n = 3) with different letters in the same column are significantly different (Two-way ANOVA; Tukey’s test, p ≤ 0.05).

Figure 6: The antifungal capacity for EOJT on solid media (disc method).
The interesting antifungal capacity of EOEJ against *A. niger*, *C. albicans*, and *F. oxysporum* may be due to their active ingredients, especially the high content of bioactive molecules in EOEJ such as muurolol, elemol, and pinene. Many studies have reported that these molecules have a strong antifungal activity, most notably, the study by Chang et al. [48], which showed that muurolol had strong antifungal activity against harmful fungi, whereas elemol was also shown to have strong antifungal activity [41]. For pinene compound, many studies have shown that this compound has significant antifungal activity. Nóbrega et al. [49], showed a significant antifungal action of $\alpha$-pinene against *Candida* spp, and Shi et al. [50], reported also significant antifungal activity of pinene against five plant pathogens including *C. gloeosporioides*, *F. proliferatum*, *A. kikuchiana*, and *Phomopsis* sp.

Many strategies had been devoted to the control of *A. flavus*, *F. oxysporum*, *A. niger*, and *C. albicans* using different types of substances, either natural or chemically synthesized. Our results are in agreement with previous

**Table 3: antifungal activity of EOEJ.**

|                  | Candida albicans ATCC 10231 | Aspergillus niger MTCC 282 | Aspergillus flavus MTCC 9606 | Fusarium oxysporum MTCC 9913 |
|------------------|-------------------------------|-----------------------------|-----------------------------|-----------------------------|
| Antifungal activity (mm) | 20.33 ± 1.15$^a$ | 0.67 ± 0.00$^a$ | 21.39 ± 0.57$^a$ | 10.75 ± 0.00$^a$ |
| MIC (mg/mL)      | 0.67 ± 0.00$^a$ | 0.00 ± 0.00$^a$ | 37.44 ± 0.28$^a$ | 6.45 ± 0.00$^a$ |
| Antifungal activity (%) | 67 ± 15$^a$ | 100 ± 0.00$^a$ | 0.00 ± 0.00$^a$ | 0.00 ± 0.00$^a$ |
| MIC (mg/mL)      | 0.00 ± 0.00$^a$ | 0.00 ± 0.00$^a$ | 0.00 ± 0.00$^a$ | 0.00 ± 0.00$^a$ |

(Mean ± SD, n = 3) with different letters in the same column are significantly different (Tow-way ANOVA; Tukey’s test, p ≤ 0.05).

**Table 4: Docking results of EOEJ in the active site of NADPH.**

|                 | Glide Gscore | Glide emodel | Glide energy |
|------------------|--------------|--------------|--------------|
| $\alpha$-Cadinol | −6.041       | −39.399      | −28.218      |
| Muurolol         | −5.956       | −40.689      | −28.987      |
| $\gamma$-Eudesmol | −5.542   | −34.926      | −26.230      |
| Elemol           | −4.538       | −34.725      | −26.235      |
| $\alpha$-Pinene  | −4.358       | −20.490      | −16.077      |

Figure 7: Principal component analysis (PCA) in the C1-C2 plane presents the correlations of antimicrobial activity of EOEJ compared to erythromycin and fluconazole. Ag: *A. niger*; Af: *A. flavus*; Fo: *F. oxysporum*; Ca: *C. albicans*; Sa: *S. aureus*; Ec: *E. coli*; Bs: *B. subtilis*; Pm: *P. mirabilis*.

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reports e.g., the study of Jemli et al. [51], which showed a substantial antifungal activity against *A. alternata*, *F. solani*, *F. oxysporum*, *V. dahlia*, and *R. solani* with a percentage of inhibition ranging from 24% to 92.1%.

Lafraxo et al. [13] showed that the essential oil of *Juniperus thurifera* leaves exhibited potent antifungal activity vs. *F. oxysporum* and *C. albicans* at a concentration of 0.095 g/mL.

**Figure 8**: 2D diagrams of ligands interactions with the active site of NADPH. (a) α-Cadinol; (b) muurolool; (c) γ-eudesmol; (d) elemol.
6. Conclusion

This study showed that *Juniperus thurifera* bark essential is rich in terpene compounds, which were extracted and characterized by GC-MS. Additionally, the *Juniperus thurifera* essential oil also had proven antimicrobial, antifungal, and antioxidant capacity. This opens up the possibility of encapsulating this essential oil through complex biotechnology applications with antibiotics to enhance their effects against pathogen resistance. However, before any potential application, further studies dealing with toxicity are warranted.

Data Availability

Data used to support the findings are included within the article.

Conflicts of Interest

The authors declare that they have no conflicts of interest.

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