Negative Effect of Camu-Camu (Myrciaria dubia) Despite High Vitamin C Content on Iron Bioavailability, Using a Caco-2 Cell Model

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INTRODUCTION

Iron deficiency anemia is a major public health problem, especially in developing countries where iron content in the diet is poor and is mostly in the form of poorly available nonheme iron [UNICEF, UNU, WHO, 2001]. In a national survey conducted in Peru in 2009 [Endes Continua, 2009], it was found that twenty percent of women of reproductive age, and thirty-seven percent of children under the age of five, suffer from iron deficiency anemia.

Peruvians often drink fruit juice or soda along with their meals. Camu-camu (Myrciara dubia), is a fruit found in the jungle of Peru with high vitamin C content (2,780 mg per 100 g) and juice is often made from this fruit [Collazos, 1996]. It is well known that vitamin C is an enhancer of nonheme iron absorption due to its high reducing capacity [Lynch & Cook, 1980]. It causes ferric reduction to ferrous and makes iron more soluble. Ferric iron reduction is important because it is mostly insoluble and poorly absorbed [Proulx & Reddy, 2007]. This study was therefore conducted to investigate whether camu-camu, as a good source of vitamin C can improve nonheme iron bioavailability from selected meals commonly consumed in Peru. We used in vitro digestion coupled with Caco-2 cell iron uptake to assess iron bioavailability. Caco-2 cells are human adenocarcinoma cell line that exhibit enterocyte-like biochemical and morphological characteristics and have been used widely for studies of bioavailability of heme and nonheme iron due to their high correlation with iron absorption studies in humans [Au & Reddy, 2000; Proulx & Reddy, 2006, 2007; Yun et al., 2004].

MATERIALS AND METHODS

Camu-camu juice

Camu-camu fruits were obtained from Iquitos, Loreto, Peru and were sent to Lima from the National Institute for Agrarian Innovation (INIA). They were processed manually as follows. The fruits were crushed and seed and peel were separated from the pulp. The pure camu-camu juice obtained was then placed in 50 mL vials and frozen at -80°C, before shipment to the United States.

Vitamin C determination

Ascorbic acid in camu-camu juice was measured using the method that is based on the reduction of Fe⁢³⁺ to Fe⁢²⁺ by ascorbic acid, followed by chelation of Fe⁢²⁺ by ferrozine as a chromogen [McGown et al., 1982].

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Nonheme iron content determination

Nonheme iron content of each meal was determined by the modified method of Torrance & Bothwell [1968] using ferrrozine instead of bathophenanthroline disulfonic acid as a chromogen in a microplate assay [Chidambaram et al., 1989].

Meals

Two meals, namely rice with lentils and wheat flour porridge which are common dishes in Peru were prepared as a regular adult portion sizes and used in the experiments. The rice and lentil meal was prepared from 110 g of rice and 70 g of lentils. These ingredients were cooked and homogenized with iron-free water (260 mL). In preparing the wheat flour porridge, wheat flour (28 g) was dissolved in water, heated to boil and it was further cooked after adding sugar (11 g) and powdered milk (17 g). Both meals were weighed and saved as small batches at -20°C. Camu-camu juice was prepared by mixing the pulp with water 1:5 as it is usually consumed as a refreshment drink in 1:5 water dilution (240 mL per serving). Three different treatments in triplicates were prepared with each meal as follows. Water (C₀) was added to the first aliquot; Two different amounts of Camu-camu juice were added to the second (C₁) and the third (C₂) sample. C₁ represents typical intake level in Peru and C₂ contains three times the vitamin C concentration in C₁. The amount of juice added (C₂) was proportional to the fraction of one serving of meal. To further understand the effect of camu-camu on iron bioavailability, 918 mg ascorbic acid was added to each of the two meals. This was equal to the amount of ascorbic acid in camu-camu juice that was used in C₂ sample. The food ingredients, required for preparing the meals, were obtained from local supermarket in Ames, Iowa. Deionized water and iron-free water were used throughout the study.

In vitro digestion

In vitro digestion of samples and iron bioavailability measurements were done as described by Proulx & Reddy [2006]. Brief description of the protocol is provided below. Aliquots of both meals (with camu-camu or with vitamin C) were subjected to in vitro digestion followed by Caco-2 cell uptake. For simulating gastric and duodenal digestion, pepsin and pancreatin/bile extract were added to the samples. Pepsin was prepared by solubilizing 0.2 g of porcine pepsin A (1:60000) in 5 mL of 0.1 mol/L HCl. Pancreatin and bile solutions were prepared by dissolving 0.05 g of porcine pancreatin (4 x USP) and 0.3 g bile extract in 25 mL of 0.1 mol/L sodium bicarbonate (NaHCO₃, 0.1 mol/L). Trace minerals were removed from the mixture of pepsin and pancreatin by treatment with Chelex-100 (BioRad), as described by Glahn et al. [1998]. Samples were weighed to deliver 200 µg of total iron, mixed with water, and adjusted to pH 2 using 0.1 mol/L HCl. Pepsin solution was added (1 mL) and incubated at 37°C with shaking at 800 rpm for 1 h. Following the pH adjustment to 6.0 with 1 mol/L NaHCO₃ solution, pancreatin and bile solution were added (5 mL), and incubation was continued for 15 min. Samples were heat-treated (4 min at 100°C) to inactivate proteolytic enzyme activity [Jovani et al., 2001] and centrifuged, and the supernatant was used for iron bioavailability experiments.

Iron bioavailability determination using the Caco-2 cell model

Bioavailability was determined using ferritin synthesis as an index of bioavailable iron in response to iron uptake by the Caco-2 cells which is common method used by other researchers [Glahn et al., 1998; Proulx & Reddy, 2006; Yun et al., 2004]. Ferritin values were normalized to cell protein concentration. All reagents for cell culture work were from Sigma Aldrich or Gibco BRL (Grand Island, NY) unless otherwise mentioned. Caco-2 cells were purchased from American Type Culture Collection (Rockville, MD) and the following experiments were conducted during passages 34–37. Cells were grown in a culture flask with Dulbecco’s Modified Eagle’s Medium (DMEM) with 10% fetal bovine serum (FBS), 1% v/v nonessential amino acids, and 1% v/v antibiotic-antimyotic solution and maintained at 37°C in an incubator with 5% CO₂. Media was changed 3 times weekly. At day 7, the cells were rinsed with Earle’s Balanced Salt Solution (EBSS), trypsinised to dissociate the cells, and centrifuged for 5 min. The cells were seeded on collagenized (Type 1 Rat tail collagen) 12-well cell culture plates (Corning Costar) at a density of 5 x 10⁴ cells/cm², for iron bioavailability experiments. The cell culture plates were maintained in the incubator with conditions similar to those of the cell culture flask. Iron bioavailability experiments were conducted 15 d post seeding. Serum-free media without FBS (0.5 mL) as suggested by earlier studies [Glahn, 1998], prepared with DMEM with 1% v/v nonessential amino acids, 1% v/v antibiotic-antimyotic solution, 10 mmol/L piperazine-N,N’-bis-[2-ethanesulfonic acid] (PIPES), hydrocortisone (4 mg/L), insulin (5 mg/L), selenium (5 µg/L), triiodothyronine (34 µg/L), and epidermal growth factor (20 µg/L), was mixed with equal volume of sample supernatants and was added to cell monolayer after rinsing the cells with Earl Balanced Salt Solution (EBSS) and then incubated for 2 h. An additional 0.5 mL of serum-free media was added followed by a further incubation for 22 h. After 24 h total incubation, cells uptake solution was removed and rinsed with EBSS. Then, cells were lysed by adding 0.5 mL of deionized water to each well and sonicated with a probe-type sonic dismembrator at lowest setting (<1 W output) for 15 s. Total cellular protein was determined in the lysates by the Bradford Coomassie Assay (Pierce Laboratories, Rockford, IL). Ferritin in the lysates was determined by radioimmunoassay (Fer-Iron II, Ramco Laboratories, Stafford, TX). Iron bioavailability was expressed as µg ferritin/µg cell protein. The relative bioavailability (RBA) of C₁ and C₂ was expressed as percentage of C₀ bioavailability. ANOVA with Tukey’s multiple comparison test was used to determine the differences among three treatments for each meal and unpaired t-test was used to compare RBA of C₁ and C₂ for each meal. The differences were considered significant at p≤0.05.

RESULTS AND DISCUSSION

Proximate analysis indicated that the vitamin C content of camu-camu juice was 638 mg/100 g juice. The vitamin C contents of the different concentrations of camu-camu used in this study were 306 mg for C₁, and 918 mg for C₂.
TABLE 1. Nutrient composition of meals.

| Meal               | Portion description | Energy (Kcal) | Protein (g) | Fat (g) | Carbohydrates (g) | Iron (mg) | Ascorbic acid (mg) |
|--------------------|---------------------|---------------|-------------|---------|-------------------|-----------|-------------------|
| Rice with lentils  | Per serving         | 638.7         | 25.39       | 0.742   | 127.6             | 8.8       | 3.08              |
|                    | Per 100 g raw mix   | 354.83        | 14.1        | 0.41    | 70.89             | 4.89      | 1.71              |
| Wheat flour porridge | Per serving       | 197.31        | 6.63        | 0       | 44.67             | 1.05      | 0.86              |
|                    | Per 100 g raw mix   | 352.34        | 11.83       | 0       | 79.76             | 1.87      | 1.54              |

TABLE 2. Effect of different concentrations of camu-camu on iron bioavailability in two meals.

| Result description                  | Diet                                      | C₀   | C₁    | C₂    | P       |
|-------------------------------------|-------------------------------------------|------|-------|-------|---------|
| Iron bioavailability (µg ferritin/µg protein) | Rice with lentils                         | 28±2 | 31±2  | 25±1  | 0.0853  |
|                                     | Wheat flour porridge                      | 124±4| 91±6  | 35±2  | <0.0001 |
| Relative bioavailability (% of control)   | Rice with lentils                         | -----| 111.7±6.7| 90.7±5.3|<0.0279 |
|                                     | Wheat flour porridge                      | -----| 73.7±4.4 | 28.1±1.5|<0.0001 |

Different superscripts in the same row are significantly different. Comparisons were made within the same meal.

The control treatment (C₀) had no camu-camu juice. Non-heme iron contents of the food samples were 14.7 mg and 3.3 mg per serving for the rice with lentils and wheat flour porridge, respectively. These values were higher than those from the USDA Nutrient Database (Table 1).

Results on the effect of camu-camu on iron bioavailability are shown in Table 2. In the rice with lentils meal, bioavailability of iron was not significantly (p>0.05) different among the three different levels of camu-camu. On the contrary, a clear trend was observed for the wheat flour porridge, in which the bioavailability was reduced as concentration of camu-camu increased. The bioavailability values for the three different levels of camu-camu concentration in the wheat flour porridge were significantly different from each other (p<0.0001). Iron bioavailability of C₀ without camu-camu was 3.5 times higher than C₁ and 1.4 times higher than C₂.

The inverse relationship between camu-camu concentration and iron bioavailability was made clearer by the RBA values (Table 2). For both meals, RBA values significantly decreased with increasing concentration of camu-camu. This is contrary to expectation considering the high vitamin C content of camu-camu (306 mg in C₁ treatments and 918 mg in C₂ treatments). Cook & Monsen [1977], reported a proportional increase in iron absorption by adding 25 to 1,000 mg of ascorbic acid to a semisynthetic meal containing 4.1 mg iron; and Diaz et al. [2003] showed that intake of 25 mg of ascorbic acid, in lemonade, twice a day for 2 weeks, doubled the absorption of iron from typical Mexican foods containing between 6.2 and 8.6 mg of iron. This unexpected observation of the inhibitory effect of camu-camu juice on iron absorption may be explained by the high polyphenols content of camu-camu. Apart from having a high content of vitamin C, camu-camu is also rich in polyphenols containing about 2.4% [Munoz et al., 2007]. Polyphenols are potent inhibitors of iron absorption because they can bind iron forming insoluble complexes [Siegenberg et al., 1991].

Comparing results for the two meals, bioavailability values were generally higher in the wheat flour porridge meal than the rice with lentils. The lower bioavailability in the rice with lentils meal (28 µg ferritin/µg protein) compared with the wheat flour (124 µg ferritin/µg protein) may be due to the combined effect of phytic acid and polyphenols associated with lentils. Since bioavailability of rice with lentils
was already low, the camu-camu juice did not show further significant reduction. On the other hand, in the wheat flour meal, iron bioavailability was reasonably higher in the control (C₀) meal, and the inhibitory effect of camu-camu juice was more pronounced due to its high polyphenol content. However, when the RBA was compared between C₁ and C₂, they were significantly different for both rice with lentil (p<0.03) and wheat porridge (p<0.0001) (Table 2).

Adding ascorbic acid similar to the concentration present in C₂, provided us some understanding of camu-camu’s inhibiting effect. Ascorbic acid had no effect which is similar to adding camu-camu with rice/lentil meal (Figure 1A), again suggesting phytic acid and polyphenols inhibition is too strong to overcome by ascorbic acid. Whereas, adding ascorbic significantly increased bioavailability by 3-fold with wheat porridge in contrast to a significant decrease with the addition of camu-camu (Figure 1B). This observation suggests that the polyphenol content of camu-camu overrides the enhancing effect of ascorbic acid in camu-camu leading to a reduction in iron bioavailability from the meals, particularly from the wheat flour porridge.

CONCLUSION

This study suggests that camu-camu juice may reduce iron bioavailability in the meal due to its extremely high polyphenols content which inhibits iron bioavailability, countering the enhancing effect of vitamin C. This might also imply that, though the fruit may not be useful in improving nonheme iron absorption it may be useful in protecting from chronic diseases by providing antioxidant effect due to its high polyphenol content.

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REFERENCES

1. Au A.P., Reddy M.B., Caco-2 cells can be used to assess human iron bioavailability from semipurified meal. J. Nutr., 2000, 130, 1329–1334.
2. Chidambaram M.V., Reddy M.B., Thompson J.L., Bates G.W., In vitro studies of iron bioavailability. Probing the concentration and oxidation – reduction of pinto bean iron with ferrous chromogens. Biol. Trace Elem. Res., 1989, 19, 25–40.
3. Collazos C., Tablas Peruanas de Composición de alimentos. Ministerio de Salud. 7th. Edition. 1996. Lima, Perú, p. 86.
4. Cook J.D., Monsen E.E., Vitamin C, the common cold, and iron absorption. Am. J. Clin. Nutr., 1977, 30, 235–241.
5. Diaz M., Rosado J.L., Allen L.H, Arams S., Garcia O.P., The efficacy of a local ascorbic acid–rich food in improving iron absorption from Mexican diets: a field study using stable isotopes. Am. J. Clin. Nutr., 2003, 78, 436–440.
6. Endes Continua. Encuesta Demográfica y de Salud Familiar on Perú: Resultados de la encuesta demográfica y de salud familiar, 2009, [Online]. Available: [http://desa.inei.gob.pe/endes/images/Expo_Jefe.pdf].
7. Glahn R.P., Lee O.A., Yeung A., Goldman M.L., Miller D.D., Caco-2 cell ferritin formation prediction nonradiolabeled food iron availability in in vitro digestion/Caco-2 cell culture model. J. Nutr., 1998, 128, 1555–1561.
8. Jovani M., Barbera R., Farre R., Martin de Aguilera E., Calcium, iron and zinc uptake from digests of infant formulas by Caco-2 cells. J. Agric. Food Chem., 2001, 49, 3480–3485.
9. Lynch S.R., Cook J.D., Interaction of vitamin C and iron. Ann. N.Y. Acad. Sci., 1980, 355, 32–44.
10. McGown E.L., Ruskov M.G., Lewis C.M., Tillotson J.A., Tissue ascorbic acid analysis using ferrozine compared with the dinitrophenylhydrazine method. Anal. Biochem., 1982, 119, 55–61.
11. Munoz A.M., Ramos-Escudero F., Alvarado-Ortiz C., Castaneda B., Evaluación de la capacidad antioxidante y contenido de compuestos fenólicos en recursos vegetales promisorios. Rev. Soc. Quim. Perú., 2007, 73, 55–61.
12. Proulx A.K., Reddy M.B., Iron bioavailability of hemoglobin from soy root nodules using a Caco-2 cell culture model. J. Agric. Food Chem., 2006, 54, 1518–1522.
13. Proulx A.K., Reddy M.B., Fermentation and lactic acid addition enhance iron bioavailability of maize. J. Agric. Food Chem., 2007, 55, 2749–2754.
14. Siegenberg D., Baynes R.D., Bothwell T.H., Macfarlane B.J., Lamparelli R.D., Car N.G., MacPhail P., Schmidt U., Tal A y F Mayet, Ascorbic acid prevents the dose-dependent inhibitory effects of polyphenols and the phytates on nonheme-iron absorption. Am. J. Clin. Nutr., 1991, 53, 537–541.
15. Torrance J.D., Bothwell T.H., A simple technique for measuring storage iron concentrations in formalinised liver samples. S. Afr. J. Med. Sci., 1968, 33, 9–11.
16. United Nations Children’s Fund (UNICEF), United Nations University (UNU), World Health Organization (WHO, 2001) on Iron Deficiency Anaemia. Assessment, prevention and control. A guide for programme managers. WHO/NHD/01.3. [Online]. Available at: [http://www.who.int].
17. USDA National Nutrient Data Base for Standard Reference. Nutrient Laboratory Data. Agricultural Research Service. United States Department of Agriculture (USDA) on Nutrient Laboratory Data. [Online]. Available at: [http://www.nal.usda.gov].
18. Yun S., Habicht J.P., Miller D.D., Glahn R.P., An in vitro digestion/Caco-2 cell culture system accurately predicts the effects of ascorbic acid and polyphenolic compounds on iron bioavailability in humans. J. Nutr., 2004, 134, 2717–2721.

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