Cyclization

Sustainable Synthesis of Chiral Tetrahydrofurans through the Selective Dehydration of Pentoses

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Abstract: l-Arabinose is an abundant resource available as a waste product of the sugar beet industry. Through use of a hydrazone-based strategy, l-arabinose was selectively dehydrated to form a chiral tetrahydrofuran on a multi-gram scale without the need for protecting groups. This approach was extended to other biomass-derived reducing sugars and the mechanism of the key cyclization investigated. This methodology was applied to the synthesis of a range of functionalized chiral tetrahydrofurans, as well as a formal synthesis of 3R-3-hydroxymuscarine.

The effective use of biomass, and in particular that generated as waste,[1] is essential to reduce the global dependence on petrochemical resources for the manufacture of valuable compounds, fuels and materials.[2] The majority of biomass is made up of carbohydrates, which are an abundant source of pentoses and hexoses.[3] For example, the refinement of sugar beet generates beet pulp as a major waste product, and this is a rich source of l-arabinose.[4] A variety of techniques has been developed to convert these biomass resources into valuable small molecules, such as the dehydration of pentoses under forcing acidic conditions to give furfural (Scheme 1), which can then be converted into various alcohols, alkenes, and heterocycles.[5] However, the majority of compounds prepared from pentoses and hexoses in this fashion are either achiral[6] or racemic mixtures where the stereochemistry of the chiral precursors is lost.[7] Using these products as intermediates in the synthesis of more complex targets may therefore require the reintroduction of stereocenters using asymmetric catalysis[8] or resolutions.[9]

The tetrahydrofuran (THF) is a privileged scaffold within medicinal chemistry[10] and the stereoselective synthesis of chiral THFs has been a major area of recent research.[11] An attractive approach is to utilize the inherent chirality present in single isomer biomass-derived carbohydrates.[12] However, existing methods often require the selective conversion of a primary alcohol into an alkyl sulfonate or halide[13] and/or the use of protecting groups,[14] both of which are detrimental to the economy of a synthetic route.[15] Herein we describe the application of N,N-dimethylhydrazine[16] for the selective dehydration of biomass-derived reducing sugars to prepare chiral THFs under mildly acidic conditions (Scheme 1).[17]

Treating l-arabinose 1a with N,N-dimethylhydrazine and Amberlyst 15 acidic resin in methanol at room temperature gave hydrazone 2a in 99% yield (Table 1, entry 1). Stirring hydrazone 2a in methanol at 40 °C for 16 h with 20 mol% TFA resulted in 100% conversion of 2a. Analysis of the crude 1H NMR spectrum indicated the formation of THF 3a as a 75:25 mixture of diastereoisomers and purification by flash column chromatography gave a mixture of the two stereoisomers in 67% yield. The reaction was scaled up from a 6.7 mmol scale to a 104 mmol scale without any significant drop in yield, giving 11.9 g of THF 3a. The major diastereoisomer was isolated by recrystallization and the stereochemistry was confirmed by...
single-crystal X-ray diffraction (Figure 1). Both steps were conducted in a sustainable solvent\(^{[18]}\) (methanol) without the need for either pre-drying of the solvent or for a drying agent in the reaction.

The same reaction conditions were used to prepare the enantiomeric THF \(\text{ent-3a}\) from \(\text{D-ribose}\) (Table 1, entry 2) in a 58% yield over two steps. It is noteworthy that the diastereoselectivity of this reaction was comparable with that observed for the cyclization of arabinose-derived hydrazone \(\text{2a}\). The methodology was also extended to \(\text{D-lyxose}\) (Table 1, entry 3), with the corresponding hydrazone prepared in 98% yield. The TFA-mediated cyclization step gave THF \(\text{3b}\) in 66% yield as a 55:45 mixture of diastereoisomers. THF \(\text{3b}\) could also be prepared from \(\text{D-xylose}\) in 61% yield over two steps, again as a 55:45 mixture of diastereoisomers (entry 4). This is a particularly important result as \(\text{D-xylose}\) is one of the major components of biomass.\(^{[3]}\) Xylose is naturally available in both enantiomers and using \(\text{L-xylose}\) it was possible to access \(\text{ent-3b}\) in a comparable yield (entry 5). The methodology was extended to deoxy sugar \(\text{L-rhamnose}\), another constituent of sugar beet pulp, to give THF \(\text{3c}\) in 69% yield as a 60:40 mixture of diastereoisomers (entry 6).

Recrystallization of hydrazone \(\text{3a}\) yielded the major \(\text{anti-dia-stereoisomer}\) in high purity. Reducing hydrazone \(\text{anti-3a}\) using hydrogen, a palladium catalyst and \(\text{Boc}_2\text{O}\) gave carbamate \(\text{4}\) in 60% yield as a single stereoisomer (Scheme 2).

Treatment of THF \(\text{3a}\) (d.r. = 75:25) with Amberlyst\(^*\) 15 acidic resin in water at room temperature resulted in rapid hydrolysis of the hydrazone to give hydrolyzed product \(\text{5}\) (Scheme 3).\(^{[20]}\)

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Table 1. Two-step synthesis of THFs 3 from sugars 1.

| Entry | Sugar 1 | Step 1 yield [%] | THF 3[^b] | Step 2 yield [%] | d.r.[^c] |
|-------|---------|------------------|-----------|-----------------|---------|
| 1     | L-arabinose \(\text{1a}\) | 99 | - | 67 (66)[^d] | 75:25 |
| 2     | D-ribose \(\text{1b}\) | 98 | - | 59 | 75:25 |
| 3     | D-lyxose \(\text{1c}\) | 98 | - | 66 | 55:45 |
| 4     | D-xylose \(\text{1d}\) | not isolated | - | 61[^e] | 55:45 |
| 5     | L-xylose \(\text{ent-1d}\) | not isolated | - | 57[^e] | 55:45 |
| 6     | L-rhamnose \(\text{1e}\) | 99 | - | 69 | 60:40 |

[^a]: Reagents and Conditions: \(\text{NH}_2\text{NMe}_2\) (2.0 equiv), Amberlyst\(^*\) 15, MeOH, 24 h, RT. [^b]: Reaction conducted on a 6.0–6.7 mmol scale unless otherwise stated. [^c]: Determined by analysis of the crude \(^1\)H NMR spectra. [^d]: Reaction conducted using 20.0 g (104 mmol) of hydrazone \(\text{2a}\). [^e]: Yield over two steps from xylose.

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**Figure 1.** ORTEP of the asymmetric unit in the crystal structure of hydrazone \(\text{anti-3a}\). The thermal ellipsoids are shown at a 50% probability level. Only hydrogen atoms belonging to the cyclic core are shown for clarity.\(^{[19]}\)

**Scheme 2.** Reduction of hydrazone \(\text{anti-3a}\).
Reduction of compound 5 with NaBH₄ in methanol gave triol 6 as an 85:15 mixture of diastereoisomers in 98% yield over two steps from hydrazone 3a. Reductive amination of intermediate 5 using n-butylamine, acetic acid, and hydrogen/palladium, followed by trapping of the intermediate amine with Boc₂O, gave carbamate 7 in 65% yield from hydrazone 3a as an 80:20 mixture of diastereoisomers. Compound 5 was also converted to alkene 8 using trimethyl phosphonoacetate in 73% yield over two steps with excellent E-selectivity. Finally, treating compound 5 with Amberlyst 15 in methanol resulted in the formation of dimethyl acetal 9 in 74% yield over two steps with excellent E-selectivity. Finally, treating compound 5 with Amberlyst 15 in methanol resulted in the formation of dimethyl acetal 9 in 74% yield over two steps from 3a as a 65:35 mixture of stereoisomers.

The hydrolysis/reduction sequence was also applied to the hydrazones 3b and 3c, which gave the corresponding triols 10 and 11 in 90% and 93% yield respectively. L-Rhamnose-derived triol 11 is a late-stage intermediate in Fleet's synthesis of 3R,3-hydroxymuscarine 12. Triol 11 was previously prepared from L-rhamnose using stoichiometric bromine, trifluoromethanesulfonic anhydride, and lithium aluminium hydride, so our route represents a less hazardous and more sustainable alternative.

A plausible reaction mechanism for the cyclization of hydrazone 2a is proposed in Scheme 4. The N,N-dialkylhydrazone group of 2a could promote the acid-mediated elimination of the adjacent hydroxyl to give vinyldiazenium intermediate 13. Cyclization of this intermediate would give THF 3a as either an anti- or syn-diastereoisomer. Resubmission of an isomerically pure sample of 3a to the reaction conditions resulted in the same 75:25 mixture of anti- and syn-diastereoisomers that was observed in the original reaction, which suggests that the diastereoselectivity is under thermodynamic control. Conducting the reaction in [D₄]MeOH did not result in detectable incorporation of deuterium adjacent to the hydrazone, indicating that epimerization occurs through a reversible ring closure rather than via a vinyldiazenium intermediate. The proposed mechanism is also consistent with the observation that hydrazones 2a and 2b converge to THF 3a and ent-3a with the same diastereoselectivity (Table 1, entries 1 and 2), as the two reactions would proceed through enantiomeric vinyldiazenium intermediates. Without TFA present no cyclization of 1a was observed.

In a preliminary study, the extension of this approach to hexoses was explored (Scheme 5). Hydrazone 14, formed from α-galactose, was subjected to the TFA-mediated cyclization conditions. This gave a 60:40 mixture of THF 15 and tetrahydropyran 16 in 53% isolated yield, with both heterocycles formed as single stereoisomers.
In summary, we have developed an efficient multi-gram approach to low-molecular weight chiral molecules from biomass feedstock sources. This route allows access to a range of THF products without the need for protecting groups, including a formal synthesis of 3R,3-hydroxymuscarine. On the basis of experimental evidence, we have proposed a reaction mechanism for the key cyclization involving a vinylidienazirine intermediate.

Experimental Section

Experimental procedures, 1H and 13C NMR spectra, characterization data for all compounds and crystallographic data for anti-3a are available in the Supporting Information.

A mixture of hydrazone 2a (20.0 g, 104 mmol) in MeOH (210 mL, 0.5 M) was treated with TFA (1.5 mL, 24 g, 20 mol%) at room temperature and the reaction stirred at 40 °C for 16 h. The reaction was quenched with aq. sat. NaHCO3 and concentrated in vacuo to give the crude THF (anti-syn = 75:25). This was purified by flash column chromatography (80:100 hexane:acetone) to give THF 3a (11.9 g, 68.3 mmol, 66%, anti-syn = 75:25).

**anti-3a**: Isolated as a single stereoisomer following recrystallization from boiling CPME; white crystalline solid; m.p. = 65–67 °C; Rf = 0.33 (1:1 acetone:hexane); \( \nu_{\text{max}} \) (film/cm\(^{-1} \)) = 3415 s, 1585 s, 1467 s, 1445 s; 1H NMR (600 MHz; [D\(_6\)]MeOH) 6.51 (1H, d, \( J = 6.6, N = CH \)), 4.23–4.18 (2H, m, N = CHCH\(_2\)CH\(_2\)CH\(_2\)); 4.08 (1H, dd, \( J = 9.6, 4.9, OCH\)), 4.02 (1H, dd, \( J = 7.3, 5.1, N = CHCH\)), 3.76–3.72 (1H, m, OCH\(_2\)), 2.79 (6H, s, N(CH\(_2\)))\(_3\); 13C NMR (150 MHz; [D\(_6\)]MeOH) 135.6 (C=N), 82.5 (CHCH\(_2\)), 76.5 (N = CHCH\(_2\)), 73.9 (OCH\(_2\)), 72.4 (CH\(_2\)CHCH\(_2\)), 42.8 (N(CH\(_2\))\(_3\)); HRMS (EI\(^+\)) found [M + H\(^+\)]\(^+\) = 174.0979; \( C_9H_{12}N_2O_2 \) requires 174.0999; [\( \alpha \)](20 °C) = +85.8 (anti-3a, MeOH, C = 1.4).

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