Biosorption chrome (Cr) and dyes using biosorbert in the modified tea bag

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Abstract

Biosorption has been known as a potential method in reducing heavy metals and dyes in wastewater, including chrome (Cr) and dyes contents of the wastewater from the batik industry. Straw and spent mushroom compost (SMC) are potential biosorbents due to cheap, abundant, and easily obtained. However, the effectivity of the biosorbents is not only depended upon the type of material but also their surface area. In this study, a modified tea bag was used to increase the surface area of the straw and spent mushroom compost to absorb Cr and dyes from the batik industry wastewater. The optimum of biosorbent ratio and pH in Cr absorption and dyes decolorization was measured. This experiment was conducted using Spilt Plot Design. The results showed that the highest Cr absorption was found at 0.0050 mg/g, and the percentage of decolorization was 68.92% in the biosorbent ratio of 3:1 and pH 5. Biosorbent packed in the tea bags modification was effective removes Cr and dyes in the batik industry wastewater.

Keywords: batik, dyes, heavy metals, natural biosorbent, wastewater

Introduction

Indonesia has been known for its various batik industries. However, inappropriate management of batik wastewater treatment causes pollution in water bodies around the batik industries in Indonesia. For example, Cr concentration in wastewater from the Sokaraja batik center in Banyumas Regency (Central Java) reaches 0.231 mg.L\(^{-1}\) [1] (Lestari et al. 2015) of which exceeds the threshold set by the Indonesian government No. 82 of 2001 (< 0.05 mg.L\(^{-1}\)).

Among the available methods of wastewater treatment management, the biological method is one of the cheapest methods that can be used in batik wastewater treatment. In this method, the biosorption mechanism through the application of biosorbents has been reported effective in removing heavy metals and dyes from various wastewater (Okuo et al. 2006). Biosorption is sufficient to absorb heavy metals and dyes in wastewater due to their high absorption rate and broad spectrum (Veglio & Beolchini 1997, Azmat et al. 2007). Various types of biosorbent have been used to reduce various pollutants, particularly heavy metals. These include powdered Triplochiton scleroxylon (Akissi et al. 2013), activated carbon activated by Fe\(_3\)O\(_4\) (Abdel-fattah et al. 2014), Sargassum filipendula alginate (Bertagnolli et al. 2014), Spirulina platensis (Kwak et al. 2015), peanut hull (Owalude & Tella 2016), Nauclea diderrichii seed (Omorogie et al. 2016), and Sargassum hystrix (Ghasemi et al. 2016).
In the study of Cr absorption using biowaste, a combination of various biosorbents provides higher effectivity in Cr removal than using a single biosorbent. For example, the application of S. cinereum and SMC of Pleurotus ostreatus mixture (3:1) was reported more effective in absorbing Cr than individual application (Lestari 2016). A combination of Annona squamosa and Aspergillus niger seeds (Babu & Preetha 2014), apple seeds and A. niger (Taylor et al. 2015) as biosorbents were also more effective than a single application of biosorbent.

A straw of Oryza sativa and SMC of P. ostreatus contains lignin and cellulose compounds that have the potential to absorb Cr and dyes. A surface area (Joo et al. 2011) and the active group in lignin and cellulose in the form of O-H and C = O play a role in Cr binding (Lestari 2018). Placing biosorbents in a container is necessary to increase the surface area of the biosorbents. Among the containers, tea bags are useful because they have pores that can be passed by wastewater and are easily separated from wastewater. However, the tea bag method has weakness due to biosorbent clots formation (Lestari 2016); therefore, it is necessary to modify this method through various absorbents ratio experiment. The objective of this study is to examine the optimum pH and ratio of O. sativa straw and SMC of P. ostreatus mixture in the tea bag in Cr and dyes removal from batik wastewater.

Materials and methods

Samples and experimental design

Wastewater samples were collected from the batik industrial center in Kauman Sub Village, Sokaraja Kulon Village, Sokaraja Subdistrict, Banyumas Regency, Central Java. In the experiment, the initial pH of the wastewater was set to 5, 6, 7, 8, and 9.

This experiment was conducted using Spilt Plot Design. The main plot was the mixture of O. sativa straw and SMC of P. ostreatus, and a subplot was pH. Five different ratios (1:0, 3:1, 1:1, 1:3, and 0:1) of mixed biosorbents were used in this study. A total of 300 mg of biosorbents was used in each tea bag (6 x 6 cm) with the ratio of mixture refers to the experimental treatment.

Cr and dyes adsorption assay

The absorption experiment was conducted in the Erlenmeyer 250 mL, of which each Erlenmeyer was filled 100 mL of batik wastewater, and one packed biosorbent in the tea bag with the composition and particle size according to treatment. All the Erlenmeyer were incubated in the shaker incubator at a 175 rpm and a temperature 25ºC for 1 hour.

Absorption capacity was calculated by the following formula:

\[
q = \frac{V (C_o - C_{eq})}{m}
\]

- \(q\) = absorption capacity (mg.g\(^{-1}\))
- \(V\) = volume of solution (L)
- \(m\) = biosorbent weight (g)
- \(C_o\) = initial concentration (mg.L\(^{-1}\))
- \(C_{eq}\) = final concentration (mg.L\(^{-1}\))

A percentage of decolorization was measured at each interval treatment using the spectrophotometric method. A total of 5 mL of each treated wastewater was centrifuged at 5,000 rpm for 10 min. The supernatant was further measured by a spectrophotometer at 645 nm. The decolorization percentage was calculated using the following formula:
The results showed that there was a decrease in the concentration of Cr in the batik wastewater (Fig. 1). The Cr concentration before being treated with *O. sativa* straw and SMC of *P. ostreatus* biosorbents range from 0.0179 mg.L\(^{-1}\) to 0.0282 mg.L\(^{-1}\) but after the biosorbents treatment range from 0.0012 mg.L\(^{-1}\) to 0.0045 mg.L\(^{-1}\). The highest Cr absorption capacity (0.0050 ± 0.003 mg.g\(^{-1}\)) was found in *O. sativa* straw and SMC of *P. ostreatus* (3:1) at pH 5 (Fig. 1) with 75.72% absorption efficiency from the initial concentration of 0.0219 to 0.0016 mg.L\(^{-1}\). The lowest Cr absorption capacity (0.0029 ± 0.001 mg.g\(^{-1}\)) was found in *O. sativa* straw and SMC of *P. ostreatus* (3:1) at pH 9 (Fig. 1.) with 62.00% of absorption efficiency from the initial concentration of 0.0219 to 0.0041 mg.L\(^{-1}\).

In the decolorization assay, the highest decolorization capacity (68.92%) was found in the ratio of *O. sativa* straw : SMC of *P. ostreatus* (3:1), and the lowest (44.81%) was found in the ratio of *O. sativa* straw : SMC of *P. ostreatus* (1:3) (Fig. 2).
Figure 2. Decolorization capacity of mixture of *O. sativa* straw and SMC of *P. ostreatus* in a modified tea bag

Discussion

This study showed that a mixture of *O. sativa* straw and SMC of *P. ostreatus* could absorb about 0.0167 to 0.0237 mg L$^{-1}$ of Cr from the batik wastewater (Fig. 1). The Cr absorption capacity in all *O. sativa* straw and *P. ostreatus* SMC mixture ratios of the biosorbent decreased when the wastewater acidity increased (from pH of 5 to 9) (Fig. 1). This data indicates that the biosorbent mixture activity is useful in acidic conditions. A previous study using a mixture *S. cinereum* and SMC of *P. ostreatus* also showed that metal absorption increased along with the decreasing of pH value (Lestari 2016). Although the absorption capacity is primarily related to the hydroxyl and carboxyl functional groups (active sites in the metal absorption capacity) in the straw and SMC mixture, however, the acid condition is probably related to the mycelium activity in absorbing Cr. Fungi usually produce organic acids when absorbing heavy metals, and the heavy metals oxidation process is more stable in acidic condition (Fomina et al. 2007). For example, *Penicillium janthinellum* F-13 was reported producing citric acid during aluminum toxicity reduction (Zhang et al. 2002), and *P. simplicissimum* (current name: *P. janthinellum*) produced citric acid during adsorption of Zn(II) (Franz et al. 1991).

Biosorption process is one of the passive removal mechanisms of heavy metal ions (Gupta 2015). The process occurs when heavy metal ions bind to the cell wall in two different ways, i.e. (1) ion exchange in which monovalent and divalent ions such as Na$^+$, Mg$^{2+}$ and Ca$^{2+}$ in the cell wall replaced by heavy metal ion, and (2) complex formation between heavy metal ion with functional formations such as carbonyl, amino, hydroxyl, phosphate, and hydroxyl-carboxyl located on the cell wall. The active process coincides in line with the metal ion consumption and intracellular accumulation of the metal ion.

The ANOVA analysis showed that there was no significant difference between Cr absorption with the ratio of biosorbents mixture and the wastewater pH, or the interaction between the two biosorbents to the Cr absorption. It showed that the Cr absorption by the biosorbents mixture was not affected by the ratio of the biosorbent mixture and wastewater pH. However, the absorption capacity influenced by the volume of biosorbent mixture in the tea bag. It is probably related to the number of the active binding site. The more volume of
biosorbents in the tea bag increase the number of –OH and –COOH of the active binding sites. The SMC of P. ostreatus also contains cellulose and hemicellulose (Kartikasari et al. 2012, Sorta et al. 2012) of which provides more binding sites to absorb Cr from the wastewater.

Dyes decolorization activity by the mixture of O. sativa straw and SMC of P. ostreatus also showed a significant result (Fig. 2). The current decolorization activity (44.81-68.92%) showed that a mixture of O. sativa straw and SMC of P. ostreatus is more effective than a previous study that employed a mixture S. cinereum and SMC of P. ostreatus (39.92%) in the decolorization of the batik wastewater (Lestari 2018). It is also related to the activity of the mycelium of P. ostreatus in the SMC. The presence of mycelium was significant increases the decolorization of dyes in the batik wastewater because the fungal mycelium is natural dyes absorbent. The mechanism of fungal mycelium in absorbing dyes is usually through enzymatic and non-enzymatic mechanisms (Murugesan & Kalaichelvan 2003). The enzymatic process generally through laccase, lignin peroxidase, and manganese peroxidase enzymes activity, while non-enzymatic process via cell wall binding sites of fungi which contains glycoproteins, chitin, and glucans (Lu et al. 2017).

Conclusion

The optimum ratio of O. sativa straw and the SMC of P. ostreatus packaged in the modified tea bag to absorb Cr in the batik wastewater was 3 : 1 at pH 5.

Conflict of Interest

The authors state no conflict of interest from this manuscript.

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References

Abdel-fattah TM, Mahmoud ME, Osmam MM, Ahmed SB. 2014. Magnetically active biosorbent for chromium species removal from aqueous media. J. Environ. Sci. Heal. 49, 1064-1076.

Akissi LCK, Adouby K, Yao B, Boa D. 2013. Batch and column studies for Cadmium (II) removal using sawdust from Triplochiton scleroxylon. International Journal of Engineering Science and Technology 5(1), 31-40.

Azmat R, Uzma, Uddin F. 2007. Biosorption of toxic metals from solid sewage sludge by marine green algae. Asian J. Plant Sci. 6(1), 42-45.

Babu E, Preetha B. 2014. Kinetics and equilibrium studies on biosorption of chromium (VI) by mixed biosorbents. International Journal of Chem Tech Research 6(12), 4927-4933.

Bertagnolli C, Da Silva MGC, Guibal E. 2014. Chromium biosorption using the residue of alginate extraction from Sargassum filipendula. Chem. Eng. J. 237, 362-371.

Fomina M, Charnock J, Bowen AD, Gadd GM. 2007. X-ray absorption spectroscopy (XAS) of toxic metal mineral transformations by fungi. Environ. Microbiol 9, 308-321.

Franz A, Burgstaller W, Schinner F. 1991. Leaching with Penicillium simplicissimum: influence of metals and buffers on proton extrusion and citric acid production. Appl. Environ. Microbiol. 57: 769-774.

Ghasemi FF, Dobarakarad S, Raeisi A, Esmaili A, Mohammadi MJ, Keshtkar M, Nasab SG, Soleimani F. 2016. Data on Fe (II) biosorption onto Sargassum hystrix algae obtained
from the Persian Gulf in Bushehr port, Iran. Data Brief, 9, 823-827. DOI: 10.1016/j.dib.2016.10.018.

Gupta VK. 2015. Biosorbents for remediation of heavy metals: current status and their future prospects. Environmental Engineering Research 20(1), 1-18. DOI: 10.4491/eer.2015.018

Joo JH, Hussein KA, Hassan SHA. 2011. Bacteria and fungi as alternatives for remediation of water resources polluting heavy metals. Korean J. Soil Sci. Fert 44(4), 600-614. DOI: 10.7745/KJSSF.2011.44.4.600

Kartikasari TH, Lestari S, Dewi RS. 2012. Adsorpsi Zn dan dekolorisasi limbah batik menggunakan limbah baglog Pleurotus ostreatus dengan sistem inkubasi dan volume limbah batik berbeda. Biosfera 29(3), 167-173. (In Indonesian language)

Kwak HK, Kim MK, Lee JY, Yun H, Kim MH, Park YH, Lee KH. 2015. Preparation of bead-type biosorbent from water-soluble Spirulina platensis extracts for chromium (VI) removal. Algal Research 7, 92-99. DOI: 10.1016/j.algal.2014.12.006

Lestari S. 2016. Biosorpsi krom total dalam limbah cair batik dengan biosorben yang dikemas dalam kantung teh celup. Biosfera 33(2), 71-75. (In Indonesian language)

Lestari S. 2018. Biosorpsi krom pada limbah cair batik menggunakan biosorben dan toksisitasnya terhadap ikan Nila (Oreochromis niloticus Linnaeus, 1758). Disertasi. Gadjah Mada University. (In Indonesian language)

Lestari S, Sudarmadji, Tandjung SD, Santosa SJ. 2015. Kajian kualitas air kali Wangan yang tercemar limbah cair batik. pp. 553–556, in Seminar Nasional Pengelolaan Sumber Daya Alam dan Lingkungan. (in Indonesian language)

Lu T, Zhang Q, Yao S. 2017. Removal of dyes from wastewater by growing fungal pellets in a semi-continuous mode. Front. Chem. Sci. Eng. 11(3): 338–345. DOI 10.1007/s11705-017-1644-0

Murugesan K, Kalaichelvan PT. 2003. Synthetic dye decolourization by white rot fungi. Indian Journal of Experimental Biology 41, 1076-1087.

Okuo JM, Sanni SB, Aigbedion SU. 2006. Selective biosorption of heavy metal ions from aqueous solutions by pre-treated Nigerian fresh water algae. Trends Appl. Sci. Res. 1(1), 83-90.

Omorogie MO, Babalola JO, Unuabonah EI, Song W, Gong JR. 2016. Efficient chromium abstraction from aqueous solution using a low-cost biosorbent: Nauclea diderrichii seed biomass waste. J. Saudi Chem. Soc. 20(1), 49-57.

Owalude SO, Tella AC. 2016. Removal of hexavalent chromium from aqueous solutions by adsorption on modified groundnut hull. Beni-Suef Univ. J. Basic Appl. Sci., 5(4), 377-388. DOI: 10.1016/j.bjbas.2016.11.005

Sorta RT, Lestari S, Dewi RS. 2012. Dekolorisasi beberapa macam limbah cair batik menggunakan limbah baglog Pleurotus ostreatus dengan waktu inkubasi berbeda. Biosfera 29(3), 136-140. (In Indonesian language)

Taylor P, Saravanan A, Kumar PS, Preetha B. 2015. Optimization of process parameters for the removal of chromium (VI) and nickel (II) from aqueous solutions by mixed biosorbents (custard apple seeds and Aspergillus niger) using response surface methodology. Desalination and Water Treatment 57, 14530-14543. DOI: 10.1080/19443994.2015.1064034

Veglio F, Beolchini F. 1997. Removal of metals by biosorption: a review. Hydrometallurgy 44(3), 301-316. DOI: 10.1016/S0304-386X(96)00059-X

Zhang D, Duine JA, Kawai F. 2002. The extremely high Al resistance of Penicillium janthinellum F13 is not caused by internal or external sequestration of Al. Biometals 15: 167-174.