Primary antioxidant free radical scavenging and redox signaling pathways in higher plant cells

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Antioxidants in plant cells mainly include glutathione, ascorbate, tocopherol, proline, betaine and others, which are also information-rich redox buffers and important redox signaling components that interact with cellular compartments. As an unfortunate consequence of aerobic life for higher plants, reactive oxygen species (ROS) are formed by partial reduction of molecular oxygen. The above enzymatic and non-enzymatic antioxidants in higher plant cells can protect their cells from oxidative damage by scavenging ROS. In addition to crucial roles in defense system and as enzyme cofactors, antioxidants influence higher plant growth and development by modifying processes from mitosis and cell elongation to senescence and death. Most importantly, they provide essential information on cellular redox state, and regulate gene expression associated with biotic and abiotic stress responses to optimize defense and survival. An overview of the literature is presented in terms of primary antioxidant free radical scavenging and redox signaling in plant cells. Special attention is given to ROS and ROS-antioxidant interaction as a metabolic interface for different types of signals derived from metabolisms and from the changing environment. This interaction regulates the appropriate induction of acclimation processes or execution of cell death programs, which are the two essential directions for higher plant cells.

Key words: Plant cells; Antioxidant; Reactive oxygen species (ROS); ROS-antioxidant interacting interface; Soil-plant interface; Signaling

1. Introduction

Higher plants survive in a constantly fluctuating environment, which has driven the evolution of a highly flexible metabolism and growth (basic growth curve of slow-fast-slow) and development necessary for their sessile lifestyle [1-3, 17-19]. The cellular state is defined as the total active level that regulates cells to adapt to their surroundings in a narrow sense [4, 10, 13-16]. Redox state is the important part of the above cellular state and one of the most important indicators for evaluating the above state [4, 9, 14-16]. Primary intracellular plant antioxidants expression (or levels or location) are closely related to their metabolic state and is responding to constantly fluctuating environment [5-9]. Practically, this phenomenon is also quite similar to animals in most cases. Because higher plants are sessile, they develop into a series of pathways at different levels that combat with environmental stress, which produces more ROS [10-16]. These pathways include the phloem respiratory pathway, enzymatic and non-enzymatic pathways, corresponding responsive-gene regulation and anatomical ways [17-21]. General stress includes drought, salinity, low temperature, UV-B and others. Currently, in contrast to the situation in the natural world, the detailed dissection of the regulatory networks that govern higher plants’ responses to abiotic insults and their biotic stress aspects have been studied almost exclusively in controlled environments where a single challenge has been applied [4-12, 19-21]. Plant cell metabolism must be highly regulated in order to allow effective integration of a diverse spectrum of biosynthetic pathways that are reductive in nature [13-15, 61-64], in which the regulation does not completely avoid photodynamic or reductive activation of molecular oxygen to produce ROS, particularly superoxide, \( \text{H}_2\text{O}_2 \) and singlet oxygen [15-17]. However, in many cases, the production of ROS is genetically programmed, induced during the course of development and by environmental fluctuations, and has complex downstream effects on both primary and secondary metabolism [18-20, 59-64]. Higher plant cells produce ROS, particularly superoxide and \( \text{H}_2\text{O}_2 \) and, as second messengers in many
processes associated with plant growth and development [21-26, 62]. Moreover, one of the major ways in which higher plants transmit information concerning changes in the changing environment is via the production of bursts of superoxide at the plasma membrane [27, 28].

Situations which provoke enhanced ROS production have in the past been categorized under the heading of oxidative stress, which in itself is a negative term implying a harmful process, when in fact it is probably in many cases quite the opposite, enhanced oxidation being an essential component of the repertoire of signals that higher plants use to make appropriate adjustments of gene expression and cell structure in response to environmental and developmental cues [29, 30, 61-64]. Rather than involving simple signaling cassettes, emerging concepts suggest that the relationship between metabolism and redox state is complex and subtle [31, 63]. Related functioning processes for main antioxidants and redox signaling in plant cells were summarized in Figure 1. This article covered several important aspects of antioxidants and redox signaling in higher plants, including biological roles of main antioxidants, ROS and redox signaling. Antioxidants and redox sensing mechanisms, and a network for redox signaling.

2. Biological Roles of Main Antioxidants in Plant Cells

2.1. α-tocopherol

α-tocopherol, found in green parts of plants scavenges lipid peroxo radicals through the concerted action of other antioxidants [31, 45]. Further, tocopherols were also known to protect lipids and other membrane components by physically quenching and chemically reacting with O₂ in chloroplasts, thus protecting the structure and function of PSII [32, 46]. Researchers reported a two-fold increase in α-tocopherol in turf grass under water stress [33, 47].

In plant cells, the most important reducing substrate for H₂O₂ removal is ascorbic acid [38, 39, 62]. A direct protective role for ascorbic acid has also been demonstrated in rice, where partial protection against damage caused by a release from flooding conditions was provided by the prior addition of ascorbic acid [40]. α-Tocopherols (vitamin E) are lipophilic antioxidants synthesized by all plants. α-Tocopherols interact with the polyunsaturated acyl groups of lipids, stabilize membranes, and scavenge and quench various reactive oxygen species (ROS) and lipid soluble byproducts of oxidative stress [41, 42].

Singlet oxygen quenching by tocopherols is highly efficient, and it is estimated that a single α-tocopherol molecule can neutralize up to 120 singlet oxygen molecules in vitro before being degraded [43]. Because of their chromanol ring structure, tocopherols are capable of donating a single electron to form the resonance-stabilized tocopheroxyl radical [44]. α-Tocopherols also function as recyclable chain reaction terminators of polyunsaturated fatty acid (PUFA) radicals generated by lipid oxidation [45]. α-Tocopherols scavenge lipid peroxo radicals and yield a tocopheroxyl radical that can be recycled back to the corresponding α-tocopherol by reacting with ascorbate or other antioxidants [46]. α -tocopherols are major lipid soluble antioxidant present in the PUFA-enriched membranes of chloroplasts and are proposed to be an essential component of the plastid antioxidant network. The most attributed function of tocopherols is their involvement in various mechanisms in protecting PUFAs from oxidation [47]. ROS generated as by-products of photosynthesis and metabolism are potential sources of lipid peroxidation in plant cells. α-tocopherol levels increase in photosynthetic plant tissues in response to a variety of abiotic stresses [48]. α-tocopherols scavenge and quench various ROS and lipid oxidation products, stabilize membranes, and modulate signal transduction [48, 49].

Synthesis of low-molecular-weight antioxidants,
such as α-tocopherol, has been reported in drought-stressed plants [49]. Oxidative stress activates the expression of genes responsible for the synthesis of tocopherols in higher plants [49, 50]. Antioxidants including α-tocopherol and ascorbic acid have been reported to increase following triazole treatment in tomato and these may have a role in protecting membranes from oxidative damage, thus contributing to chilling tolerance [51]. Triazole increases the levels of antioxidants and antioxidant enzymes in wheat [52]. Photosynthetic apparatus and membrane could be affected by water stress. Alpha-tocopherol is a lipid-soluble antioxidant associated with biological membrane of cells, especially the membrane of photosynthetic apparatus [8, 10, 14, 21]. Research has shown that water deficiency may result in an increase of tocopherol concentration in plant tissues [50, 51]. Some evidence implied that tocopherol content of soybean leaves was increased as the amount of rainfall decreased [52]. This is consistent with the reports of [53], which showed that subjecting spinach to water deficit increased the content of tocopherol in the leaves. Based on the studies of 10 different grass species under water stress, the researcher found that drought stress led to an increase of 1 to 3-fold of tocopherol concentration in 9 out of 10 species [53, 60]. They pointed out that the species with a high tolerance of stress are defended through tocopherol. Moreover, highly significant correlations were observed between stress tolerance and tocopherol concentration (the precursor of α-tocopherol; Spearman’s rank correlation coefficient r = 0.731).

2.2. Ascorbic acid (AA)

AA is one of the most extensively studied anti-oxidants and has been detected in the majority of plant cell types, organelles and apoplast [34, 35]. AA is synthesized in the mitochondria and is transported to the other cell components through a proton–electrochemical gradient or through facilitated diffusion. Further, AA also has been implicated in regulation of cell elongation [36, 49]. In the ascorbate-glutathione cycle, two molecules of AA are utilized by APX to reduce H\textsubscript{2}O\textsubscript{2} to water with concomitant generation of monodehydroascorbate. Monodehydro-ascorbate is a radical with a short life time and can disproportionate into dehydroascorbate and ascorbic acid. The electron donor is usually NADPH and catalyzed by monodehydro-ascorbate reductase or ferredoxin in water-water cycle in the chloroplasts [37].

A continuous oxidative assault on plants during drought stress has led to the presence of an arsenal enzymatic and non enzymatic plant antioxidant defenses to counter the phenomenon of oxidative stress in plants [54]. AA is an important antioxidant, which reacts not only with H\textsubscript{2}O\textsubscript{2} but also with O\textsubscript{2}, OH and lipid hydroperoxidases [46, 55]. AA is water soluble and also has an additional role in protecting or regenerating oxidized carotenoids or tocopherols [55, 56]. Water stress resulted in significant increase in antioxidant AA concentration in turf grass [57].

AA showed a reduction under drought stress in maize and wheat, suggesting its vital involvement in deciding the oxidative response [58]. Some reports reported a decrease in the level of antioxidants including AA with increase in stress intensity in wheat. AA can also directly scavenge 1O\textsubscript{2}, O\textsubscript{2} and •OH and regenerate tocopherol from tocopheroxyl radicals, thus providing membrane protection [58, 59]. AA also acts as a co-factor of violaxanthin de-epoxidase, thus sustaining dissipation of excess excitation energy [59, 60]. Antioxidants such as AA and glutathione are involved in the neutralization of secondary products of ROS reactions [45, 61].

Ascorbate (vitamin C) occurs in all plant tissues, usually being higher in photosynthetic cells and meristems (and some fruits). About 30 to 40% of the total ascorbate is in the chloroplast and stromal concentrations as high as 50 mM have been reported [62]. It is highest in the mature leaf, where the chloroplasts are fully developed and the chlorophyll levels are highest. Although it has been determined that D-glucose is the precursor of L-α-Asp, the synthetic pathway has not been totally understood [61, 63]. AA has effects on many physiological processes including the regulation of growth, differentiation and metabolism of plants. A fundamental role of AA in the plant defense system is to protect metabolic processes against H\textsubscript{2}O\textsubscript{2} and other toxic derivatives of oxygen. Acting essentially as a reductant and reacting with and scavenging many types of free radicals, AA reacts non-enzymatically with superoxide, hydrogen peroxide, and singlet oxygen. It can react indirectly by regenerating -tocopherol or in the synthesis of zeaxanthin in the xanthophyll cycle. Therefore, AA influences many enzyme activities, and minimizes the damage caused by oxidative process through synergic function with other antioxidants [60-63].

2.3. Reduced glutathione

Glutathione is a tripeptide (α-glutamyl cysteinylglycine), which has been detected virtually in all cell compartments such as cytosol, chloroplasts, endoplasmic reticulum, vacuoles and mitochondria [64]. Glutathione is the major source of non-protein thiols in most plant cells. The chemical reactivity of the thiol group of glutathione makes it particularly suitable to serve a broad range of biochemical functions in all organisms. The nucleophilic nature of the thiol group also is important in the formation of mercaptide bonds
with metals and for reacting with selected electrophiles. This reactivity along with the relative stability and high water solubility of GSH makes it an ideal biochemical to protect plants against stress including oxidative stress, heavy metals and certain exogenous and endogenous organic chemicals [63, 64].

Glutathione takes part in the control of H₂O₂ levels [56, 63]. The change in the ratio of its reduced (GSH) to oxidized (GSSG) form during the degradation of H₂O₂ is important in certain redox signaling pathways [64]. It has been suggested that the GSH/GSSG ratio, indicative of the cellular redox balance, may be involved in ROS perception [63, 64]. Reduced glutathione (GSH) acts as an antioxidant and is involved directly in the reduction of most active oxygen radicals generated due to stress. There was a report reporting that glutathione, an antioxidant helped to withstand oxidative stress in transgenic lines of tobacco [39, 63].

3. ROS and Redox Signaling

Higher plants, as other aerobic organisms, require oxygen for the efficient production of energy [40]. During the reduction of O₂ to H₂O₂, reactive oxygen species (ROS), namely superoxide radical (O₂•⁻), hydrogen peroxide (H₂O₂), and hydroxyl radical (OH• ) can be formed [41, 42]. Most cellular compartments in higher plants have the potential to become a source of ROS. Environmental stresses that limit CO₂ fixation, such as drought and salt stress, ozone and high or low temperatures, reduce the NADP⁺ regeneration by the Calvin cycle, consequently, the photosynthetic electron transport chain is over-reduced, producing superoxide radicals and singlet oxygen in the chloroplasts [43-45, 56, 59]. To prevent over-reduction of the electron transport chain under conditions that limit CO₂ fixation, higher plants evolved the photorespiratory pathway to regenerate NADP⁺ [47, 49]. As part of the photorespiratory pathway, H₂O₂ is produced in the peroxisomes, where it can also be formed during the catabolism of lipids as a by-product of β-oxidation of fatty acids [50, 62].

Because of the highly cytotoxic and reactive nature of ROS, their accumulation must be under tight control. Higher plants possess very efficient enzymatic and non-enzymatic antioxidant defence systems that allow scavenging of ROS and protection of plant cells from oxidative damage [51-53, 60-62]. The distinct subcellular localization and biochemical properties of antioxidant enzymes, their differential inducibility at the enzyme and gene expression level and the plethora of non-enzymatic scavengers render the antioxidant systems a very versatile and flexible unit that can control ROS accumulation temporally and spatially [54, 61-64]. The above controlled modulation of ROS levels is significant in the light of the recent evidence for a signaling capacity of ROS [55, 60-64].

Higher plants can sense, transduce, and translate the ROS signals into appropriate cellular responses, the process of which requires the presence of redox-sensitive proteins that can undergo reversible oxidation/reduction and may switch ‘on’ and ‘off’ depending upon the cellular redox state [56, 61]. ROS can oxidize the redox-sensitive proteins directly or indirectly via the ubiquitous redox-sensitive molecules, such as glutathione (GSH) or thioredoxins, which control the cellular redox state in higher plants [57, 63, 65, 66]. Redox-sensitive metabolic enzymes may directly modulate corresponding cellular metabolism, whereas redox-sensitive signaling proteins execute their function via downstream signaling components, such as kinases, phosphatases, and transcription factors [58, 59, 62]. Currently, two molecular mechanisms of redox-sensitive regulation of protein function prevail in living organisms [23-25, 36, 60-64]. Signaling mediated by ROS involves hetero-trimeric G-proteins [29] and protein phosphorylation regulated by specific MAP kinases and protein Tyr phosphatases [31, 63]. The biochemical and structural basis of kinase pathway activation by ROS remains to be established in higher plants, but thiol oxidation possibly plays a key role [37, 46, 62]. The best-characterized redox signal transduction system in higher plants is the stromal ferredoxin-thioredoxin system, which functions in the regulation of photosynthetic carbon metabolism. Signal transmission involves disulfide-thiol conversion in target enzymes and is probably achieved by a light-induced decrease in the thioredoxin redox potential from about -0.26V to about -0.36V [27, 36, 61]. Thiol groups are likely important in other types of redox signal transduction, including ROS sensing by receptor kinases, such as ETR1 [48, 60]. ROS in higher plants must utilize and/or interfere with other signaling pathways or molecules, forming a signaling network [61, 64]. Increasing evidence shows that higher plant hormones are positioned downstream of the ROS signal. H₂O₂ induces accumulation of stress hormones, such as salicylic acid (SA) and ethylene [31, 36]. Tobacco plants exposed to ozone accumulate ABA and induction of the PDF1.21 gene by paraquat is impaired in Arabidopsis mutants insensitive to jasmonates (JA) and ethylene [42, 46]. Higher plant hormones are not only located downstream of the ROS signal, but ROS themselves are also secondary messengers in many hormone signaling pathways [16, 19, 64-66]. No doubt, feedback or feedforward interactions may conceivably occur between different hormones and ROS [23, 28, 39, 57].
4. Concluding Remarks

Besides exacerbating cellular damage, ROS can act as ubiquitous signal molecules in higher plants. ROS are a central component in stress responses and the level of ROS determines the type of response. Antioxidants and ROS are an important interacting system with different functions in higher plants, which ensures themselves a highly flexible organism [35, 39, 58]. Redox signal transduction is a universal characteristic of aerobic life honed through evolution under natural and selecting pressure to balance information from metabolism and the changing environment [62-66]. Higher plant cells can be considered as a series of interconnecting compartments with different antioxidant buffering capacities determined by differences in synthesis, transport and/or degradation. There is a set of discrete locations where signaling is controlled (or buffered) independently in higher plant cells, which permits redox-sensitive signal transduction to occur in locations such as the appoplast, the thylakoid, and the endoplasmic reticulum, whereas other highly buffered spaces have a much higher threshold for ROS signals. Both oxidants and antioxidants fulfill signaling roles to provide information on higher plant health, particularly in terms of robustness for defense, using kinase-dependent and in-dependent pathways that are initiated by redox-sensitive receptors modulated by thiol status. Antioxidants are not passive bystanders in this crosstalk, but rather function as key signaling components that constitute a dynamic metabolic interface between higher plant cell stress perception and physiological responses. Increasing current data suggest that glutathione is a key arbiter of the intracellular redox potential, and ascorbate is particularly influential in setting threshold for apoplastic and cytoplasmic signaling [57, 60-66]. Differential antioxidant concentrations between compartments make antioxidant-driven vectorial signaling through processes such as ascorbate-driven electron transport or futile cycles. There can be no doubt that transgenic plants will be invaluable in assessing the precise role that main antioxidants and ROS play in the functional network that controls stress tolerance. Approximate redox potentials and intracellular concentrations of main redox couples in plant cells (modified from Foyer and Noctor, 2005) and stress-related signal transduction via wider scope of tested plants. Basic research leading to the characterization of tightly regulated stress-inducible promoters that are also responsive to appropriate tissue-specific regulation and endogenous developmental programmes is likely to be critical in improving the overall field performance of transgenic crops. The future will determine more precisely how ascorbate, glutathione, and tocopherol are involved in initiating and controlling redox signal transduction and how they trigger the gene expression of other related responses to optimize survival strategies. In addition, other problems are how antioxidants coordinate growth and development of higher plants in a constantly changing environment, how redox signaling is linked with hormonal regulation, nutrient status and redox potential of higher plants, and how their redox signaling is cooperated with inter-and intracellular signaling, transport capacity, developmental and environmental cues to maintain an appropriate dynamic homeostasis for stress tolerance and efficient survival.

Table 1 Approximate redox potentials and intracellular concentrations of main redox couples in plant cells (modified from Foyer and Noctor, 2005)

| Redox Couples * | Redox Potentials (V) | Concentration Range (μ M)b |
|-----------------|----------------------|-----------------------------|
| O₂/H₂O | +0.82 | 200-300 (O₂) |
| O₂/O₂⁻ | -0.30 | <0.001 (O₂⁻) |
| O₂⁻ / H₂O₂ | +0.94 | 1-100 (H₂O₂) |
| H₂O₂/OH⁺ | +0.54 | Negligible (OH⁺) |
| OH⁻ / H₂O | +2.20 | — |
| DHA/ASC | -0.10 | 10, 000-20, 000 |
| GSSG/GSH | -0.24 | 2, 000-5, 000 |
| TRXred/TRXox | -0.33 | 10-100 |
| NAD(P)H/NAD(P) | -0.32 | 200-500 |
| Fdred/Fdox | -0.42 | 10-100 |

a ASC, ascorbate; DHA, dehydroascorbate; Fd red, oxidized ferredoxin; Fd red, reduced ferredoxin; GSH, glutathione; GSSG, glutathione disulfide; TRX red, oxidized thioredoxin; TRX oxidized thioredoxin; TRX red, reduced thioredoxin.

b High intracellular concentrations of antioxidants prevent indiscriminate oxidation of key cellular components by maintaining oxidant concentrations low. In addition to the abundant pools of ascorbate and glutathione and the battery of peroxide processing enzymes, numerous other compounds can act as chemical antioxidants, including tocopherols, quinones, lipoic acid, carotenoids, and flavonoids.

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Conflict of interest
The authors have declared that no conflict of interest exists.

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