Sorption and leaching characteristics of pesticides in volcanic ash soils of Jeju Island, Korea

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Abstract
It is important to evaluate leaching behavior in agricultural soils to prevent the pollution of groundwater by pesticides. We identified the distribution coefficients (Kd) of ten pesticides with different physicochemical properties and compared their leaching characteristics using wick lysimeters from three distinct soil types on Jeju Island. The Kd values varied by pesticide and soil, but were within the range of 1.2 to 4231 L kg⁻¹. Based on the European standard (Kd < 10 L kg⁻¹), six pesticides (alachlor, ethoprophos, carbofuran, napropamide, tebuconazole, and etridiazole) were mobile in at least one tested soil, and their soil organic carbon affinity was ≤ 5.811. This value differed greatly from the other pesticides (16.533 and higher). The solubility of the six mobile pesticides was ≥ 32 mg L⁻¹, which substantially differed from the other pesticides (≤ 0.71 mg L⁻¹). Thus, we conclude that our mobility assessment, which is based on Kd values, can be used to predict the leaching of pesticides in the volcanic ash soils of Jeju Island. The use of pesticides should be strictly controlled to reduce the possibility of groundwater contamination.

Keywords: Distribution coefficient, Leaching potential, Lysimeter, Organic carbon content, Pesticides, Volcanic ash soils

Introduction
Pesticides released into the soil undergo volatilization, absorption by plants, sorption, and decomposition within the soil. Leaching occurs when pesticides that have not been adsorbed or decomposed during the process of moving through the soil, flow downwards into lower soil layers with percolating water. As the pesticides leached from the soil move through the rock layer, they are likely to cause groundwater pollution if relevant filtration processes are absent [1]. To prevent the pollution of groundwater by pesticides, evaluating their leaching behavior in agricultural soils is important. There are increased concerns about groundwater pollution caused by pesticides in islands.

The mean annual precipitation of Jeju Island, Korea is approximately 2000 mm, which is greater than inland areas of Korea. Due to the characteristics of the volcanic geographical features of the island, at least 40% of the rainfall flows into the groundwater [2]. Further, an annual average of approximately 10,000 tons of pesticides has been applied on Jeju Island since 2000 [3]. Pesticide application time coincides with the heaviest rainfall period and as a result, there is a high possibility that the pesticides will reach the groundwater. The use of pesticides is unavoidable, so evaluating the probability for pesticides to flow into the groundwater and developing standards is necessary. Since groundwater pollution caused by pesticides are mainly dependent on soil characteristics, understanding the leaching behavior of pesticides in soils is required to regulate the use of pesticides for preventing groundwater pollution in Jeju Island.
During the leaching of pesticides, the rate of sorption varies with soil characteristics (e.g., organic matter content within the soil and pH) and pesticide characteristics (e.g., solubility and vapor pressure). The sorption equilibrium of pesticides is the most critical information to aid in understanding the concentration of pesticides that can be accommodated by variable organic matter contents of volcanic ash soils in Jeju Island. Linear and Freundlich equations are frequently used for evaluation, and the Freundlich equation demonstrated a higher correlation of coefficient values in the volcanic ash soils of Jeju Island [4, 5].

The distribution coefficient ($K_d$) is a value indicating the proportion of pesticides distributed on the soil mineral phase and in the solution, thus, the $K_d$ values increases as the amount of sorption increases [6]. Soil texture and structure, organic matter, depth to groundwater, geology, temperature, light, moisture, bacteria, soil pH, etc., all affect the principal process of pesticides occurring in the environment. Sorption of pesticides is affected by the properties of the pesticide, including the solubility and ionization of the pesticide, and the physicochemical properties of the soil. Among soil characteristics, soil pH, clay content (soil charges), and organic carbon content are the most important factors affecting the sorption of pesticides. Soil pH and clay content are involved in sorption of ionic pesticides, and organic carbon content is primarily involved in sorption of non-ionic pesticides [4, 5, 7]. The $K_d$ values is used as a basic parameter to investigate the possibility of applied pesticides not being sorbed into the soil and to predict the mobility of pesticides into groundwater [7–9]. The mobility of pesticides is determined by a $K_d$ cutoff value, below which pesticides are deemed mobile. This cutoff value is 5 L kg$^{-1}$ in the United States of America (USA) and 10 L kg$^{-1}$ in Europe (Germany, Denmark, and the Netherlands) [10]. The cutoff values must be used depending on the properties of soils, such as the organic carbon content and pH of the soil, and the solubility of the pesticide. Organic carbon content of the soil and the solubility of the pesticides have the strongest influence on the $K_d$ values. The volcanic ash soils of Jeju Island generally have high organic carbon content, although it can vary widely depending on soil type. Therefore, the combination of $K_d$ values with soil organic carbon content of Jeju Island can be used to predict pesticide leaching.

Studies on the leaching characteristics of pesticides on farmlands have predominantly used lysimeters, which were first implemented approximately 300 years ago [11]. Wick and pan lysimeters have most commonly been used for recent pesticide leaching experiments. Francaviglia et al. [12] suggested experimental guidelines for lysimeter studies. Selecting appropriate lysimeters is critical for pesticide leaching experiments under natural soil conditions. Zhu et al. [13] reported that the recovery rate of the wick lysimeter was over double of the rate for the pan lysimeter. Schmidt and Lin [14] indicated an excellent recovery rate of the wick lysimeter following measurements for the recovery rate of bromide. In addition, the permeability coefficient of the soils used with the wick lysimeter was identical to the soils from the sampling area, which demonstrates the possibility to conduct an experiment using the same conditions as a natural soil in a given study area [15–17]. We therefore used the wick lysimeter to predict leaching characteristics in our study based upon favorable results from previous studies.

Previous studies on the leaching potential of pesticides on Jeju Island have used indices (e.g., groundwater ubiquity score, retardation factor, and attenuation factor) that are based on the physicochemical properties of the soil and the sorption properties of pesticides [18, 19]. Hyun et al. [19] evaluated the possibility of groundwater contamination of specific pesticides in volcanic ash soil in Jeju Island using the Groundwater Ubiquity Score (GUS). They classified pesticides with a GUS $>$ 2.8 (GUS index is a leacher) such as alachlor, metolachlor, bromacil, ethoprophos, carbofuran, and metalaxyl as groundwater contaminants. These pesticides had a $K_d$ value of 5 L kg$^{-1}$ or less and their solubility was high from 200 to 8400 mg L$^{-1}$. Contrarily, in Hawai’i volcanic ash soil, the transport of selected pesticides was predicted with the GUS values and the leaching of pesticides was studied [20]. In this result, it was predicted that the fungicide trifloxystrobin would not be leached with GUS $<$ 1.8 (GUS index is a non-leacher). However, trifloxystrobin was the most mobile pesticide among the pesticide tested in field leaching test. The results suggested that the prediction of pesticide leaching at the laboratory lever using GUS values might not be consistent with leaching of pesticides in the actual field. However, an assessment to verify pesticide mobility using lysimeters has not been conducted until now in the volcanic ash soils of Jeju Island.

To bridge this gap, the present study calculated the $K_d$ values of ten pesticides (three fungicides, four insecticides and three herbicides) that vary in physicochemical properties from a three-soil series of representative volcanic ash soils in Jeju Island (Donghong, Jeju, and Pyungdae) that contain differing organic carbon content. Further, this study aimed to provide data that can predict the probability of groundwater inflow for local pesticides by investigating the leaching rate of each pesticide using a wick lysimeter, and identifying their $K_d$ values and leaching characteristics.
Materials and methods
Study area
Jeju Island (N33°06′–34°00′, E126°08–126°58′) was formed by volcanic activity that occurred between approximately 1.88 Ma to the Holocene [21, 22]. The area is covered by pyroclastic materials and lava in layers released from the area’s 360 monogenetic volcanoes and polygenetic composite volcanoes, including Hallasan Mountain (1950 m above sea level) [22]. The island has a mild oceanic climate, and the average annual temperature is 15 °C without any variation among areas. However, the average annual rainfall (2082 mm) exhibits considerable variations. Normally, as the elevation increases by 100 m, the rainfall increases by 100 to 250 mm. The average rainfall of the eastern and southern areas is 2000 mm, whereas average precipitation is relatively lower in the western coastal areas (1100 mm) [23]. In addition, due to the high permeability of soils and rocks, there are no permanent streams or rivers. Thus, the majority of the drinking water, agricultural water and water for industries are supplied by groundwater [21, 23].

The major parent material of the soils of Jeju Island is basalt and some soils that originate from trachyte and trachyctic andesite [24]. Andisols, which comprise 80% of the soils, are distributed within the central and southeastern parts of the island and are accompanied by heavy rainfall [25]. Alfisols and Mollisols are widely distributed throughout the western and northern coastal areas and the middle mountainous areas, and these are accompanied by relatively less rainfall. Inceptisols and Ultisols with andic characteristics are distributed in areas of high elevation [26, 27]. The soils of Jeju are classified by color: black soils, very dark brown soils, dark brown soils, and brown forest soils (Fig. 1) [25]. Most of the soils that are distributed among the lava terraces are used to cultivate farm products (e.g., tangerines, carrots, potatoes, and radish), and some soils are used as grasslands.

Soil sampling and analysis of physical and chemical properties
Soil samples from three types of soil (Donghong, Jeju, and Pyungdae) with varying properties, as detailed in the Jeju soil map [28] (Fig. 1), were collected from the top 0–20 cm. As the presence of pesticides is expected in farmland soils, the soil samples were collected from non-farmlands with consideration for the study location’s agricultural use and distribution areas. According to the Taxonomical Classification of Korean Soils [26], Donghong soil (dark brown) is classified as Mollis Paleudalfs and is located near the Wondang Peak of
Jocheon-eup. Jeju soil (very dark brown) is classified as Andic Palehumults with a silt loam texture and is located in Haean-dong, Jeju-si. Pyungdae soil (black) is classified as Acrudoxic Melanudand with a silt loam texture and is located in Gyorae-ri, jeju-si (Fig. 1). The collected soils were air-dried and sieved with a 2 mm sieve to analyze their physicochemical properties, sorption and leaching.

Among the chemical properties of the soil, we analyzed the pH and organic carbon contents. The pH of the soil was measured using an Orion Star A211 pH meter (Thermo Scientific, UK) with a 1:5 ratio of soil and distilled water. Organic carbon contents were analyzed using the Walkley and Black wet digestion method [29].

The following physical soil properties were analyzed: particle density, bulk density, porosity, and hydraulic conductivity. The particle density of the soil was measured using the pycnometer method [30], and the bulk density was measured by the core method [31]. Porosity was estimated through the particle and bulk density. The hydraulic conductivity was measured through the saturated hydraulic conductivity using the falling head method (Daiki, Japan) based on Darcy’s law [32]. The saturated hydraulic conductivity is expressed in m d⁻¹.

Pesticides
The ten types of pesticides used in this study (fungicides: Etridiazole, Tebuconazole, and Tolclofos-methyl; insecticides: Carbofuran, Cypermethrin, Ethoprophos, and Fluazinam; and herbicides: Alachlor, Napropamide, and Pendimethalin) were randomly selected to test soil sorption and leaching. These pesticides vary in their physicochemical properties, including solubility, and are commonly used in citrus orchards and agricultural lands on Jeju Island. The pesticides were purchased from Chem Service (PA, USA) in technical grades. Their general physicochemical properties and IUPAC names are outlined in Table 1 [33].

Simple evaluation of distribution coefficient
To determine the distribution coefficient (Kd) for the three soil series of representative volcanic ash soils in Jeju Island (Donghong, Jeju, and Pyungdae), we used single solution concentrations (10 mg L⁻¹) of ten pesticides. The Kd were simply calculated as the amount of adsorbed pesticide for single solution concentration (10 mg L⁻¹) of the pesticides on the three soil series. They were calculated as $K_d$ (L kg⁻¹) = $S C$⁻¹, where $S$ is the concentration of pesticides adsorbed to the soil and $C$ is the equilibrium concentration within the soil solution [19].

The stock solution was prepared by using methanol to obtain 1000 mg L⁻¹. After we diluted the stock solution to obtain a 10 mg L⁻¹ concentration for each pesticide, and this solution was prepared with water with 0.01 M CaCl₂ as a background electrolyte.

Sorption experiment was performed through the batch equilibration technique. Soil (5 g) and the pesticide solution (25 mL, soil: solution = 1:5) were placed into a centrifuge tube, shaken at 200 rpm for 24 h at 25 °C, and then centrifuged for 30 min at 2500 rpm to separate the soil and solution. About 0.2 g of NaCl was added to 5 mL of the supernatant of the centrifuged solution, and then 5 mL of a solvent in which hexane and ethyl acetate were mixed in a ratio of 1:1 was added, and the mixture was shaken vigorously for 1 min. After the mixed solution settled for 30 min, the components distributed to the organic solvents were analyzed through gas chromatography (GC) (Hewlett Packard, HP6890 Series II, USA). We used an Ultra-2 capillary column (Cross-linked 5% phenyl methyl silicone, 25 m × 0.32 mm ID × 0.52 μM) and either an electron capture detector (ECD) or nitrogen phosphorus detector (NPD) for GC analysis. In this condition, limit of detection (LOD) and limit of quantification (LOQ) of ten pesticides were 0.04–0.07 ng and 0.13–0.23 μg mL⁻¹, respectively. The correlation coefficients (R²) of linearity ranged from 0.9784 to 0.9999.

When calculating the amount of adsorbed pesticide, we considered that the pesticide was adsorbed to the soil, excluding the residual of the supernatant. We performed QA/QC for the GC determination and a blank soil used as matrix. The standard calibration curve was prepared by using four standard solutions with different concentrations (1, 2.5, 5, and 10 m L⁻¹) and then performed the same method as the sample analysis for the blank test without adding soil.

Soil organic carbon affinity of pesticide
Soil organic carbon affinity (SOCA, the relationship between the organic carbon content of the soil and the $K_d$ of pesticides) of pesticides can be a major factor for the sorption of pesticides in soils with varying organic content in Jeju Island soil. Therefore, in this study, the SOCA of each pesticide was calculated as the slope of the regression equation through simple linear regression analysis of the relationship between the $K_d$ value of each pesticide and the organic carbon content of the soil.

Lysimeter construction
The lysimeter experimental area was created within a vinyl greenhouse, where wind and sunlight can naturally enter, and rainfall is artificially controlled. To create the lysimeter experimental area, a basic framework to install the lysimeter was constructed, and the lysimeter was secured after adjusting its level. A wick lysimeter was used with a dimension of 30 × 30 × 5 cm (width × length × depth) in this experiment as
described in Fig. 2 [13, 14]. A tube for the fiberglass wick (2 cm i.d.) was used to improve the recovery efficiency, and the length of the wick within the lysimeter was set to 30 cm to maintain natural soil conditions [14]. On top of the secured lysimeter, a round plastic container (80 cm i.d.) was used to fill the soil to ensure that the lysimeter was at the center of the container. The lysimeter was filled with the three types of soil after they passed through the 2.0 mm sieves, which removed any gravel.
To fill the soil within the lysimeter, firstly, a water-sampling environment was created with round gravels from the coast that were not polluted with pesticides. A sand layer, 2 cm thick, covered the gravels, and each sand grain had a diameter of 0.06-2 mm. The experimental soil was spread out in a thickness of approximately 1 cm to ensure that as much of wick of the glass fiber was exposed to the soil. Next, the experimental soils were filled to a depth of 30 cm (i.e., average soil depth on Jeju Island) and to verify whether the lysimeter was stabilized and operated correctly, 20–40 mm of artificial rain was supplied over a period of three to 5 days for a total of 60 days. During the stabilizing period, any lost soil was replenished in the experimental area. To confirm whether the soil was stabilized, the physical properties of the soil at the time of extraction (bulk density and porosity) and the end of the leaching experiment were compared. The test plot of the lysimeter consisted of three repeated trials for the three soils, and the locations of the test plots were decided through a completely randomized design.

**Tracer and pesticides treatment**

The lysimeter leaching experiment for the pesticides within the soil was performed through the bromide tracer method [34–36]. The treatment of the bromide was conducted with 100 kg ha$^{-1}$ based on KBr and 6.7 g m$^{-2}$ of bromide. The lysimeters for the pesticides were treated with double the standard concentration, considering the actual application amount of the pesticides. The applied concentration levels are outlined in Table 2.

**Artificial rainfall and water received**

The artificial rainfall was derived from a small sprinkler during the process of soil stabilization to ensure identical rainfall levels for all experimental areas. There was an artificial rainfall event once every 7 days and this experimental design was conducted for 63 days (9 weeks), which represents the duration of heavy concentrated rainfall in the Jeju area. Leached water from the lysimeters was recovered for 3 days in a row following the artificial rainfall. The level of artificial rainfall, received water, and the recovery rate are shown in Table 3. The average artificial rainfall was 45.7 ± 2.1 mm, and the average recovery rate was 81.1%.

**Leached tracer and pesticides analysis**

The bromide leaching types, tracer, and pesticides were analyzed by identifying the leaching rate ($C/C_0$, the proportion of detected $Br^-$ during the experiment or the proportion of the process concentration ($C_0$) of each pesticide ($C$)), the pore volume (PV), and accumulated leaching rates (%)).

$Br^-$ was analyzed as a tracer through ion chromatography (Dionex DX-100, USA) after filtering the collected water sample and the standard solution with a 0.2 μm cellulose acetate filter during each experiment.

To quantify the pesticides, 500 mL of the collected water samples of each experiment were analyzed with either GC-ECD or GC-NPD. The extraction and analysis conditions were identical to those employed in the sorption experiments of pesticides mentioned above.

![Fig. 2 Schematic design of the wick lysimeter](image)
Table 2  Applied concentration of pesticides sprayed on the lysimeter

| Classification | Fungicide | Insecticide | Herbicide |
|----------------|-----------|-------------|-----------|
| Pesticide      | Etridiazole | Tebuconazole | Tolclofos-methyl | Carbofuran | Cypermethrin | Ethoprophos | Fluazinam | Alachlor | Napropamide | Pendimethalin |
| Concentration (mg m$^{-2}$) | 22.5 | 22.5 | 90.0 | 27.0 | 9.0 | 45.0 | 45.0 | 36.0 | 270.0 | 11.4 |
Results and discussion

Physical and chemical properties of soil samples

The soil used for the sorption experiment varied in its physical and chemical properties (Table 4). The organic carbon contents differed strongly among the soils: Donghong soil (dark brown) had a carbon content of 14.8 g kg⁻¹, Jeju soil (very dark brown) was 52.0 g kg⁻¹, and Pyungdae soil (black) was 137.0 g kg⁻¹. Among the physical properties, Donghong soil had the highest bulk density with 1.1 g cm⁻³, followed by Jeju soil and Pyungdae soil, but there was no great difference in the particle density. Pyungdae soil had the greatest porosity with 74.4%, followed by Jeju soil (71.1%) and Donghong soil (54.4%). Pyungdae soil had the highest hydraulic conductivity at 0.059 m d⁻¹, followed by Jeju soil (0.036 m d⁻¹) and Donghong soil (0.021 m d⁻¹).

As the soil color tends to be black on Jeju Island, there are higher amounts of organic carbon and fine sand. In addition, there is a higher porosity and lower bulk density, which leads to higher water permeability. In contrast, dark brown (Donghong) soil has a lower organic carbon content and higher bulk density, which results in lower water permeability. Very dark brown (Jeju) soil is between black (Pyungdae) soil and dark brown soil in terms of permeability [37]. On average, the organic carbon content was 16.2 g kg⁻¹ for dark brown soil, 51.9 g kg⁻¹ for very dark brown soil, and 100.1 g kg⁻¹ for black soil. The average bulk density was 1.31 g cm⁻³ for dark brown soil, 0.98 g cm⁻³ for very dark brown soil, and 0.77 g cm⁻³ for black soil [37]. The results of the physicochemical properties for the three soils were similar to the commonly known properties of each soil. Therefore, these could be used as representative soils in our experiment.

Mobility of pesticides based on the soil distribution coefficient

The soil Kd of the ten types of pesticides were calculated and their mobility was assessed by applying the European standard (Kd < 10 L kg⁻¹) (Table 5). Six pesticides (alachlor, ethoprophos, carbofuran, napropamide, tebuconazole, and etridiazole) were predicted to have mobility within at least one soil type, as the Kd values were within the range from 1.24 to 9.33 L kg⁻¹.

Based on the different types of the soils, the six pesticides were shown to be mobile within the Donghong soil that has the lowest organic carbon contents while three pesticides (alachlor, ethoprophos and carbofuran) were exposed to be mobile within the Jeju soil that has medium amounts of organic carbon content. Only one pesticide (alachlor) was shown to be mobile in the Pyungdae soil, which contains the greatest organic carbon content. Alachlor, which was predicted to be mobile in all study soils, had a similar Kd value of 10 L kg⁻¹ or less in Jeju soils as previous studies [4, 19]. Also, Donghong soil had similar Kd values (1.43–2.41%) to Hawaiian soils (1.89–2.99 L kg⁻¹) with a low organic carbon content [38]. However, the remaining four pesticides (cypermethrin, fluazinam, pendimethalin, and tolclofos-methyl) had Kd values from 26.0 to 4231 L kg⁻¹, which demonstrated no mobility in all studied soils.

The six pesticides that we predicted to be mobile and the remaining four pesticides demonstrated distinct

| Weeks | Artificial rain (mm) | Received amounts (mL) | Recovery (%) |
|-------|----------------------|-----------------------|--------------|
| 1     | 57.9±6.1             | 4820                  | 92.5         |
| 2     | 45.5±2.7             | 2811                  | 68.6         |
| 3     | 44.7±1.1             | 2751                  | 68.4         |
| 4     | 46.6±2.2             | 3691                  | 88.0         |
| 5     | 44.5±3.6             | 3228                  | 80.6         |
| 6     | 41.1±1.1             | 2773                  | 75.0         |
| 7     | 42.2±0.2             | 3091                  | 81.4         |
| 8     | 49.1±6.2             | 4182                  | 94.6         |
| 9     | 39.7±3.4             | 3300                  | 81.0         |
| Average | 45.7±2.1           | 3405                  | 81.1         |
| Total  | 411.4                |                       |              |

Table 4 Physical and chemical properties of the soil samples

| Soil series | PH (1:5) | OC (g kg⁻¹) |ρb (g cm⁻³) | Pp (g cm⁻³) | F (%) | θ (m³ m⁻³) | Ks (m d⁻¹) | Coordinates | Soil taxonomy |
|-------------|----------|-------------|-------------|-------------|------|------------|------------|-------------|---------------|
| Donghong    | 4.7      | 14.8        | 1.1         | 2.5         | 54.4 | 0.40       | 0.021      | N33°25′57.7″ E126°42′28.1″ | Mollic Paleudafs |
| Jeju        | 4.9      | 52.0        | 0.7         | 2.5         | 71.1 | 0.41       | 0.036      | N33°26′08.4″ E126°28′24.1″ | Andic Palehumults |
| Pyungdae    | 6.0      | 137.0       | 0.6         | 2.4         | 74.4 | 0.43       | 0.059      | N33°25′55.7″ E126°42′37.8″ | Acrudoxic Melanudand |

OC Organic carbon, ρb Soil bulk density, Pp Soil particle density, f Porosity, θ Volumetric water content at field capacity, Ks Hydraulic conductivity
differences in their leachability, soil organic carbon affinities (SOCA; the higher organic carbon contents, the higher Kd values), and solubility (Table 5, Additional file 1: Figs. S1 and S2). The SOCA of the six pesticides that were expected to be mobile were ≤5.81, whereas the solubility values were ≥36.0 mg L\(^{-1}\). However, the four pesticides that were predicted to be immobile contained organic carbon affinities of ≥16.5 and solubility rates of ≤1.0 mg L\(^{-1}\).

Based on these results, the Kd that reflects the organic carbon affinities and solubility of the pesticides could be used as an important standard to assess the mobility of the pesticides. In the volcanic ash soils of Jeju Island, where organic carbon content varies by region, the prediction of the mobility of pesticides according to the SOCA will be very important for not only the soil environment but also for the management and prevention of groundwater pollution.

**Leaching behavior of the tracer**
Bromide from the lysimeter displayed a leaching peak after 4 to 5 weeks for the Donghong and Jeju soils. However, for the Pyungdae soil, bromide showed a leaching peak after 3 weeks (Fig. 3). When accumulated leaching reached 100%, the pore volume (PV) of the Donghong soil was 2.00 PV, the Jeju soil was 1.65 PV, and the Pyungdae soil was 1.50 PV. Therefore, leaching from bromides occurred earliest within the Pyungdae soil, followed by the Jeju soil, and Donghong soil (Fig. 4).

Bromide commonly shows minimal sorption and moves at the same speed as water [34]. Therefore, the speed of bromide is closely related to hydraulic conductivity, and a greater hydraulic conductivity results in a faster speed. Kelly and Pomes [39] reported that a faster hydraulic conductivity led to an earlier peak of bromide during the leaching of bromide. The lysimeter experiment by Piwowarczyk et al. [40] also confirmed that the leaching of bromide was faster in soils with greater hydraulic conductivity.

Identifying the relationship between the hydraulic conductivity and the movement of bromide in the three types of soil, we also demonstrate that a greater hydraulic conductivity leads to faster movement. Further, after comparing the relationship between the PV and hydraulic conductivity at the point of 50% bromide leaching (Fig. 5), the greater water permeability displayed an inverse proportion with the PV, coinciding with the leaching peak \((r^2 = 0.986)\). This is identical to the faster movement of bromide under a greater hydraulic conductivity in natural soils, which indicates

| Pesticide      | Donghong Kd (L kg\(^{-1}\)) | Jeju Kd (L kg\(^{-1}\)) | Pyungdae Kd (L kg\(^{-1}\)) | Linear regression\(^{a}\)                                | R\(^2\)    |
|----------------|-----------------------------|---------------------------|----------------------------|--------------------------------------------------------|-----------|
| Alachlor       | 1.24 ± 0.15                 | 3.50 ± 0.38               | 8.48 ± 0.42                | Y = 0.5912x + 0.3866                                    | 0.9999    |
| Ethoprophos    | 1.80 ± 0.45                 | 3.70 ± 0.17               | 10.5 ± 2.88                | Y = 0.7277x + 0.384                                     | 0.9928    |
| Carbopuritan   | 2.70 ± 0.60                 | 4.87 ± 0.17               | 17.6 ± 2.00                | Y = 1.2694–0.2424                                      | 0.9739    |
| Napropamide    | 4.68 ± 1.72                 | 19.6 ± 1.41               | 47.2 ± 4.10                | Y = 3.4344x + 0.5136                                    | 0.9983    |
| Tebucconazole  | 5.93 ± 2.58                 | 19.5 ± 3.41               | 45.5 ± 2.56                | Y = 3.2082x + 1.8484                                    | 0.9975    |
| Etridiazole    | 9.33 ± 2.52                 | 23.5 ± 2.60               | 78.6 ± 11.8                | Y = 5.8111x – 2.3431                                    | 0.9887    |
| Tolclofo‑methyl | 26.0 ± 4.65                | 68.6 ± 2.10               | 223 ± 4.01                 | Y = 16.53x – 6.1989                                     | 0.9909    |
| Pendimethalin  | 59.5 ± 11.6                 | 193.0 ± 24.7              | 406 ± 15.5                 | Y = 27.79x + 30.733                                     | 0.9919    |
| Fluazinam      | 142 ± 35.7                  | 456 ± 41.6                | 789 ± 11.6                 | Y = 50.601x + 118.79                                    | 0.9585    |
| Cypermethrin   | 790 ± 121                   | 1523 ± 301                | 4231 ± 392                 | Y = 287.97x + 225.08                                    | 0.9905    |

\(^{a}\) Linear regressions for adsorption coefficients (Kd) and organic carbon (g kg\(^{-1}\)) of soils for each pesticide

**Fig. 3** Bromide leached fraction (C/C0) curves from lysimeters for the soil series of Donghong, Jeju, and Pyungdae over a nine-week period
that the lysimeters in this study efficiently matched the natural soil conditions of the extracted samples.

**Leaching behavior of pesticides**

We determined the leaching behavior of pesticides from a wick lysimeter with artificial rainfall once every 7 days for a total of 9 weeks. Among the six pesticides that were expected to be mobile based on the $K_d$ values, five pesticides (alachlor, carbofuran, ethoprophos, napropamide, and tebuconazole), excluding etridiazole, leached from at least one type of the experimental soil. However, the remaining five pesticides (cypermethrin, etridiazole, fluzinam, pendimethalin, and tolclofos-methyl) did not leach from any soil type during the experiment (Fig. 6).

All five pesticides (alachlor, carbofuran, ethoprophos, napropamide, and tebuconazole) leached from the Donghong soil. Only three pesticides (alachlor, ethoprophos, and carbofuran) leached from the Jeju soil, and only one pesticide (carbofuran) leached from the Pyungdae soil. This was consistent with the results of the mobility prediction based on the $K_d$ values. However, alachlor, which was expected to be mobile in the Pyungdae soil, did not leach, whereas carbofuran, which was not expected to be mobile, did leach. In addition, etridiazole, which was expected to be mobile in the Donghong soil, did not leach.

Among the five pesticides that leached, alachlor was expected to be mobile within all three types of soil. This is because its $K_d$ values was less than 10 L kg$^{-1}$ and its SOCA was the lowest among all types of pesticides (0.59), thus, potentially resulting in the most efficient leaching rate of all soils. However, the lysimeter showed that although alachlor leached slightly from the Donghong and Jeju soils and it did not leach from the Pyungdae soil. Alachlor leached after 3 weeks from the Donghong soil and created a leaching peak of 6% after 5 weeks, then leached by 0.8% from the Jeju soil after 9 weeks. The accumulated leaching from the Donghong soil was 26.0% (2.00 PV) and was 1.82% (1.65 PV) the Jeju soil. These results correspond with previous studies reporting that alachlor is more easily leached from soils with lower organic carbon contents [41, 42].

Conversely, carbofuran was only expected to leach from the Donghong and Jeju soils due to their relatively lower organic carbon contents and based on their $K_d$ values. In contrast, carbofuran leached from all soil types in the Donghong soil, and leaching began after 1 week, reached an initial peak concentration of 10% after 4 to 5 weeks, and then decreased. In the Jeju soil, leaching began after 4 weeks and reached an initial peak concentration of 5% after 7 to 9 weeks. Finally, in the Pyungdae soil, carbofuran leached slightly after 7 weeks. The accumulated leaching from the Donghong soil was 39.1% (2.00 PV), the Jeju soil was 16.1% (1.65 PV), and the Pyungdae soil was 0.34% (1.50 PV). The accumulated leaching from the Pyungdae soil was over 100 times lower than the Donghong soil. In addition, carbofuran exhibited the highest leaching rate among all leaching pesticides in this study.

The other three pesticides (ethoprophos, napropamide, tebuconazole) that leached recorded the same results predicted by the $K_d$ values. Although ethoprophos demonstrated the highest solubility (1300 mg L$^{-1}$) among the
pesticides in this study, the soil sorption experiment confirmed that it was mobile within only the Donghong and Jeju soils. Ethoprophos also leached from the Donghong and Jeju soils in the lysimeter experiment, and did not leach from the Pyungdae soil. Ethoprophos leached after 3 weeks from the Donghong soil, reached a peak leaching rate of 4.0% after 4 weeks, and then decreased. Ethoprophos showed a leaching rate of 0.2% from week seven to nine in the Jeju soil. The accumulated leaching rate was 11.6% (2.00 PV) from the Donghong soil and 0.7% (1.65 PV) from the Jeju soil. Napropamidé and tebuconazole only showed mobility in the Donghong soil during the sorption experiment, and they were slightly leached from the Donghong soil during the lysimeter experiment. Napropamide leached after 4 weeks from the Donghong soil and reached a leaching rate of 0.3% after 9 weeks; the accumulated leaching by 2.0 PV was 1.1%. A low level (0.1%) of Tebuconazole leached after 9 weeks from the
Donghong soil, and the accumulated leaching rate was 0.09% by 2.0 PV. Similar results have been reported in previous studies, which demonstrated that these three pesticides are not easy to leach from soils with higher organic carbon contents, such as the Pyungdae soil [1, 43–46].

An unexpected result from the lysimeter experiment was that ethoprophos and alachlor both showed a lower leaching than carbofuran, despite contrary predictions based on SOCA and solubility. In addition, alachlor, which was expected to be mobile within the Pyungdae soil, did not leach, whereas carbofuran leached, despite being predicted to be immobile. This could be the reason why ethoprophos and alachlor have a short half-life (approximately 14 days) that can be degraded. However, the half-life of pesticides will be different according to properties of soils and degradation mechanism. Actually, it was confirmed that alachlor was leached in other soils as well as carbofuran with a short half-life of 29 days was also leached in all studied soils. It indicates that the possibility of non-detection of both pesticides by degradation was very low. On the other hand, since the vapor pressure of ethoprophos and alachlor were relatively high with 78 and 2.9 mPa, respectively. It is possible that these two pesticides were volatilized without leaching. Based on these results, the leachability of some pesticides with short half-life in Jeju Island soil could be underestimated. Volatilization can easily occur within the Pyungdae soil due to the large field capacity and porosity, and alachlor may not have leached despite the predictions of mobility. Conversely, carbofuran had a relatively low vapor pressure of 0.08 mPa compared to the pesticides that leached, whereas its solubility was high (322 mg L⁻¹), which led to easy leaching in all experimental soils. In Donghong soil, carbofuran was found to be leached almost similar to bromide, a tracer, and moved rapidly with water. These results correspond with previous studies which confirmed that carbofuran easily leach from various types of soil [47, 48]. Based on these results, it was expected that carbofuran is easily to enter groundwater in soils with low organic carbon content such as Donghong soil. However, etridiazole was mobile within the Donghong soil with a low organic carbon content in the sorption experiment and was expected to leach in the Donghong soil when considering the SOCA (5.8) or solubility (89 mg L⁻¹). However, etridiazole did not leach from any type of soil during the lysimeter experiment, which may be due to the highest vapor pressure of any studied pesticide (1430 mPa).

**Supplementary information**

**Supplementary information** accompanies this paper at https://doi.org/10.1186/s13765-020-00555-5.

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