Phytochemical analysis and antinociceptive activity of bitter ginger (*Zingiber zerumbet*) cultivated in Manaus/Amazonas

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The present work concerns the analgesic effects of zerumbona, obtained from *Zingiber zerumbet* L. Smith cultivated in Manaus/Amazonas. The compound has been studied for decades because it has potent cytotoxic activity against liver and prostate tumor cells, colon, and breast. The plant is rich in sesquiterpenes, glycosylated flavonoids that present important pharmacological activities; standing out cytotoxic activity against neoplastic cells of cancers. The objective of this study was to test the antinociceptive activity of zerumbone (ZER) using chemical and thermal nociception models, that is, writhing test induced by acetic acid and hot plate test. ZER administered orally and intraperitoneally produced significant and dose-dependent analgesic activity against the pain, acetic acid, formalin, and capsaicin models. In addition, ZER significantly increased the dormancy of the animals in the hotplate test pain model (49-540°C). It was demonstrated that intraperitoneal (i.p.) and oral (p.o.) administration of ZER sesquiterpene in doses of 150 to 500 mg/kg i.p. and 250 to 1500 mg/kg p.o. produced significant dose-dependent inhibition of acetic acid-induced abdominal writhing, as compared to fentanyl (20 µg/kg). At the same intraperitoneally and orally doses, the ZER produced significant dose-dependent latency-time increases in the hot-plate test relative to control. It was concluded that ZER exhibits both central and peripheral antinociceptive activity, indicating it to hold therapeutic potential for the discovery of new antinociceptive drugs as an alternative for the discovery of new drugs in the control of neurogenesis. The ZER exhibited similar efficacy and strength via both oral and intraperitoneally routes. ZER is the major essential oil component, a new sesquiterpene responsible for its nociceptive effect, when compared with other antinociceptive sesquiterpenes described in literature.

Key words: *Zingiber zerumbet*, zerumbone, Zingiberacea, antinociceptive activity.

INTRODUCTION

Zerumbone (ZER) is the main component extracted from the essential oils of *Zingiber zerumbet*, a plant belonging to family Zingiberaceae, which is much used in the Malaysian traditional medicine (Dev, 1960; Koshimizu et al., 1988). However, in Amazonia, it is only used as an ornamental plant with no therapeutical ends, being
popularly known as bitter ginger. The results obtained with the essential oils extracted by steam distillation provided the lowest, yet very pure, yield percentage (99.95%) purer than that of the previous method (97%), postulated in the patent number PI0505343-9/28/11/2007. Dichlorometahne (DCM), methanol (MEOH) and ZER extracts phytochemical screening findings revealed the presence of the following compounds: terpenoids, xanthones, flavonoids, phenols, tannins and alkaloids. EM and DCM TLC indicated a stain revealed in iodine and ceric sulfate with the retention factor of approximately 0.7 cm similar to that of Zerumbona (0.8) and MEOH showing stains in iodine and UV 254 nm. Of all parts of the plant, the RZZ has been the subject of extensive chemical investigations because of its high medicinal values. Various reports have been published regarding the phytochemical content of RZZ. Attempts to isolate and identify bioactive compounds from the RZZ started since 1944 with the identification of humulene (Eddy and Leimbach, 1987); monoterpenes (Carter, 1991; Hwang and Wilcox, 1987), and zerumbone (2,6,10-cly-cloundecatrien-1-one, 2,6,9,9-tetramethyl-, (E,E,E)-) (Carter, 1991) from the essential oil of RZZ (EOZZ). It is a substance, presenting antiinflamatory (Murakami et al., 2002), antiproliferative (Takada et al., 2005), antimicrobial (Abdul et al., 2008), antibacterial (Kitayama et al., 2001) and antitumoral (Murakami et al., 2004) activities, being recommended for the treatment of liver, colon and skin cancer (Sulamain et al., 2009). Zerumbone is a monocyclic sesquiterpene compound isolated from rhizomes of Z. zerumbet. Recently some scientific research on the bioactivities of zerumbone, which was identified as a major compound of Z. zerumbet, reported that zerumbone possesses many pharmacological activities, such as chemoprevention Yob et al. (2011), antiinflammatory and antiallergic activities. Somchit et al. (2005) have earlier reported on the antinociceptive profiles of AEZZ and EEZZ administered intraperitoneally into rats and assessed using the 0.6% acetic acid-induced writhing test. ZER presents high pharmacological potential, holding great relevance in the pharmaceutical industry application, and it can be utilized in natural medicine manufacturing and cosmetics formulation (Koshimizu et al., 1988). Recent studies using the rhizome essential oil of Z. zerumbet showed it to possess both central and peripheral antinociceptive activity when tested using chemical and thermal models of noiception (Takada et al., 2005) and to produce significant antinociceptive effect in all noiception models in mice when given via intraperitoneally route, presenting antinociceptive effect in the peripheral region (Perimal et al., 2011). In another recent study, ZER produced its antinociception through the activation of nitric oxide (NO)/cyclic guanosine monophosphate (cGMP)/protein kinase C (PKC)/K+ channel pathways (Lapa et al., 2003). However, it is believed that Z. zerumbet essential oils antinociceptive activity is linked to its main component, the sesquiterpene ZER. The present work verified this belief using animal models, thereby revealing its potential in the search for new analgesic drugs.

MATERIALS AND METHODS

Plant

The rhizome of Z. zerumbet (Zingiberaceae) was collected in the Tarumã, and two exsicats were forwarded to the herbarium at INPA for botanical determination. They were identified by Prof. Dr. Paul Maas (Departament of Plant Ecology and Evolucionary Biology) - herbarium at the University of Utrecht. A voucher specimen (no. 186913) was deposited at the Herbarium of the INPA.

Essential oil extraction and compound isolation

Attaining ZER from Z. zerumbet rhizomes essential oils was accomplished utilizing a steam drag distillator. Twenty of fresh and dried rhizomes were weighed on a digital scale, then ground in an electric grinder followed by hand grater. The ground rhizome was placed in a 12 L Mariott flask, coupled to a 20 L semi industrial pressure pot, and 15 L of water or more were poured into it. Temperature control (100°C) and pressure in essential oils distillation was recorded in a Tecnal digital water circulator connected to a condensator (temperature and pressure gauges coupled to a manometer adapted to the pot’s lid) linked to the Clevenger device. The graduated sorter was utilized for collecting and verifying the yields of the essential oils. The initial temperature of distillation by vaporisation was 80°C, being later reduced to 70°C and the condensation temperature was 5 to 150°C. Extraction took 4 h. The obtained essential oils were submitted to re-crystallization. The ZER was obtained through the national patent deposit process number PI0505343-9 with a purity grade of 97%.

Chemical analyses

Thin layer chromatography analyses (TLC) were conducted on a Merck silica-gel G 60 plate and developed with iodine vapors, ultraviolet radiation (254 and 366 nm), ceric sulphate and Dragendorff reagent. The chromatographic profile of the samples was also determined through sorting analytical technique by High Efficiency Liquid Chromatography (HELC) in liquid solid stationary phase. The chromatograms were analyzed using a Shimadzu chromatograph, model QP010, column μ-Bondapack CN, 100×8 mm having as mobile phase in water:metanol (80:20) at 6 mL/min and detected at 254 nm. The Hydrogen Nuclear Magnetic Ressonance spectra (1H NMR) and Carbon thirteen Nuclear Magnetic Ressonance (13C NMR were carried out at the Biotechnology Center of Amazonia). 1H NMR analyses spectra were recorded in Varian - Mercury 500 MHz spectrometer with samples dissolved in CDCl3 (deuterated chloroform) and

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Animals

For the use of animals in vivo, this project was submitted to the Committee on Ethics and Use of Animals (CEUA)/INPA, being approved by the number of opinion 005/2012, in accordance with the current norms. The animals utilized in the experiments were albino mice (Mus musculus - albinus variety) weighing between 18 and 30 g and albino rats (Rattus norvegicus - Wistar variety) (male and female) with weight ranging from 150 and 300 g acquired from the Amazon Research Institute Bioterium.

Drugs and chemicals

The following drugs were used: fentanest, dipirone, indometacine, diclorometane (DCM) and methanol (MeOH), acetic acid (Sigma Chemical), and tween 20% (Sigma Chemical). All drugs were dissolved in 0.9% saline solution. The ZER (LTQPN/COTI/INPA) was dissolved in 1% (v/v) Tween 20%. The control group received only 1% Tween 20% as a vehicle. All drugs, chemicals and ZER solutions were prepared just prior to experiments, administer orally, and intraperoneally route to mice.

Experiments

Acetic acid-induced abdominal writhing

The procedure used was similar to the one described earlier (Koster et al., 1959). Mice were pretreated orally with ZER (50, 250, 500, 1000, 1500 mg/kg, p.o.), 1 h prior to i.p. injection of 1% acetic acid (v/v). The control group received a similar volume of vehicle (0.01 mL/100 g i.p.). Fentanest (20 µg/kg s.c.), was used as the reference drug and administered 30 min before the nociceptive agent. Following the i.p. injection of acetic acid, the animals were immediately placed into a perspex chamber and the number of writhings was recorded for 15 to 120 min, starting from 10 min post injection, as test standards (Rocha and Silva, 1968).

Hot plate

The hot plate test was performed to assess the central antinociceptive properties of ZER according to the method described previously (Koster et al., 1959) with minor modifications. In this test, the hot plate (HOT PLATE F361-INSIGHT) was maintained at 50°C. Animals were placed in the perspex cylinder on the heated surface, and the latency to a discomfort reaction (licking paws or jumping) was determined before and after ZER or drug administration. The ZER (250, 500, 1000, 1500 mg/kg, p.o. and 150, 200, 220, 250, 500 mg/kg, p.i.), vehicle (saline 0.9%; tween 20%) and fentanest (20 µg/kg s.c.) were administered 30 min prior to the beginning of the experiment. Animals were observed before and 30, 60, 120, 180 and 360 min following the ZER or fentanest administration. The cut-off time was 25 s to avoid tissue injury.

Analgesimeter test

Four to five groups of mice, each one of them with about five animals, were studied. Both paws basal volume of all animals was measured, utilizing the LE 7306 Analgesimeter (Koster et al., 1959) apparatus. In the test of the induced ear edema, crôton and formalin capsaicina because of the zerumbone (1 mg ear, topical application) was evaluated for the inhibition of edema. According to the results, the zerumbone exhibited, in doses and routes tested analgesic effect in rats and mice in vivo and in vitro. The control group received 0.9% saline solution + Tween20 (10 mL/kg) and the positive received fentanest synthetic opiate (20 µg/kg) and indomethacin (25 mg/kg). One hour following gavage (v.o) or i.p administration, the edema was induced on the animals hind paws, by injecting a 1% carrageenan solution and, the same volume of 0.9% + Tween 20 saline solution in the contra-lateral paw.

Acute toxicity

The method described by Rocha and Silva (1968) and Salustiano et al. (1969) was employed. Mice were separated into ten, six-mice groups. They were fasted overnight and then were administered with the ZER in 10, 100, 1500, 2500 and 5000 mg/kg doses via intraperoneal and oral routes, while the control group only received the vehicle (0.9% saline + Tween 20). The mice were observed during 180 min for any abnormal behavior such as sedation, respiratory distress, motor impairment and hypothermia and left to be observed 24 h, 48 h and 14 days later. Activities general tests were repeated in mice injected with identical doses and via the same routes.

Statistical analysis

All findings were acquired through statistical analyses of variance (ANOVA) with mean standard deviation and significance of p<0.05. The Student-Newman-Keuls test was adopted. The ID50 (dose that produced 50% inhibition in total time of paw licking) and 95% confidence intervals (CI) values were determined by using linear regression and graphs were drawn by using GraphPad Prism 4 software.

RESULTS

Structural determination of ZER from the essential oil of rhizomes Z. zerumbet (L.) Smith. The substance designated as zerumbone obtained from the recrystallization of essential oils from Z. zerumbet rhizomes was colorless, crystalline in appearance and the physico-chemical analysis of the substance showed a melting point of 64.5 to 64.57°C. Table 1 shows the 1 H NMR spectra exhibiting olefinic hydrogen signals at δ 5.25, 6.02, 5.99 and 5.87 with similar multiplicities. The purity of zerumbone was determined by high-performance liquid chromatography (HPLC), and was shown to exceed 99.95%. 13C NMR spectra showed 15 carbon signals, as 11 hydrogenated carbons (4 CH3, 3 CH2 and 4 CH), data provided by the DEPT-135 spectrum. Z. zerumbet rhizomes essential oils analyses findings confirmed the presence of a major sesquiterpene component called Zerumbone (Figure 1), which was characterized by UV-visible spectrophotometry, high efficiency liquid chromatography (HPLC), by gas chromatography and identified through 1H and 13C NMR.
Table 1. Comparison between chemical shift signals obtained from 13C and 1HNMR spectroscopic data in the literature and zerumbone obtained from the essential oil of Z. zerumbet (L.) rhizomes.

| 13CNMR | Dai et al., (1997) | Pinheiro (2005) | Hydrogen | Dai et al. (1997); Pinheiro (2005) |
|--------|-------------------|-----------------|----------|-----------------------------------|
| C-1    | 42.2              | 42.37           | CH₂      | 1.9 (d; J=16 Hz); 2.2-2.5 (m)     | 1.9 (d; J=16 Hz); 2.17-2.36 (m) |
| C-2    | 125.0             | 124.97          | CH       | 2.17-2.36 (m)                     | 2.17-2.36 (m) |
| C-3    | 136.1             | 136.21          | C        |                                    |                                    |
| C-4    | 39.4              | 39.39           | CH₂      |                                    |                                    |
| C-5    | 24.3              | 24.34           | CH₂      |                                    |                                    |
| C-6    | 148.5             | 148.74          | CH       | 6.02 (d; J=11 Hz)                  | 6.02 (d; J=11 Hz)                  |
| C-7    | 137.8             | 137.39          | C        |                                    |                                    |
| C-8    | 203.8             | 204.27          | C=O      |                                    |                                    |
| C-9    | 127.1             | 127.11          | CH       | 5.93 (d; J=16.5 Hz)                | 5.99 (d; J=16.5 Hz)                |
| C-10   | 160.4             | 160.68          | CH       | 5.87 (d; J=16.5 Hz)                | 5.87 (d; J=16.5 Hz)                |
| C-11   | 37.8              | 3780            | C        |                                    |                                    |
| C-12   | 15.2              | 15.14           | CH₃      | 1.55 (s)                          | 1.55 (s)                          |
| C-13   | 11.7              | 117.1           | CH₃      | 1.80 (s)                          | 1.80 (s)                          |
| C-14   | 24.1              | 24.13           | CH₃      | 1.08 (s)                          | 1.07 (s)*                         |
| C-15   | 29.4              | 29.37-CH₃       | H-15     | 1.21 (s)                          | 1.21 (s)*                         |

Figure 1. Chemical structure of zerumbone.

The 97% purity grade ZER was obtained through the national patent deposit process number PI0505343-9 (Pinheiro, 2005).

Pharmacological study

The effect of ZER on writhing response in mice is as shown in Figure 2. The ZER (150, 200, 220, 250, 500 mg/kg) given i.p. caused inhibition of dose-dependent acetic acid-induced writhing, with 98% of it being observed on the dose of 220 mg/kg (n=10, p<0.05) as compared to control. Such effects were also observed in mice pretreated with Fentanest (99%, p<0.05). Furthermore, the ZER (250, 500, 1,000, 1,500 mg/kg) given v.o. route 1 h before, ZER (1,000 mg/kg) caused a significant inhibition (98%) of the acetic acid-induced pain (Figure 2). The ED₅₀ for ZER given via intraperoneal (i.p.) and oral (v.o.) routes, in this model, were 150, 6 (200-500) and (250-1,500 mg/kg), respectively. Through this test, it was established that ZER compound exerted a significant effect both via i.p. and v.o. by reducing the number of acetic acid-induced abdominal writhing contortions within 2 h. Essential oil from the rhizome of Z. zerumbet exhibited a significant antinociceptive effects on acetic acid-induced writing test in a dose-dependent manner, (Pinheiro, 2009).

From the Table 2, the administrations of the ZER (200-1500 mg/kg) v.o. and Fentanest (20 mg/kg) i.p. increased significantly the latency time to the nociceptive response in the hot plate test. In the antithermociceptive model (hyperalgesia), the treatment of animals with ZER (150-1500 mg/kg, p.i. or v.o.) altered the response to the stimulation up to 3 h following its administration (3.52 ± 0.21 min).

The application of fentanest (20 µ/kg, s.c) and ZER increased in 60% the time of reactivity (strength) of the stimulation application. Such outcomes demonstrate ZER to have central analgesic activity, similar to that presented by hypnoanalgesic drugs.

Formol-induced ZER analgesic activity test

The formalin test is one of the most widely used models...
Figure 2. Percentage of oral inhibition (v.o), obtained in the acetic acid-induced writhing test and expressed for 3 h.

Table 2. Effect of the ZER substance on the hot plate test in mice.

| Treatment | Dose (mg/kg) | Via | 30     | 60    | 90     | 120    | 180    |
|-----------|--------------|-----|--------|-------|--------|--------|--------|
| Vehicle   | -            | i.p | 4.10±0.36 | 3.82±0.35 | 4.68±0.55 | 5.70±0.36 | 7.45±0.16 |
| Fentanyl  | 20           | i.p | 3.62±0.58 | 3.62±0.61 | 8.78±0.41 | 9.38±0.57 | 10.70±0.26 |
| ZER       | 250          | v.o | 6.39±1.10 | 7.59±1.98 | 10.18±2.06 | 10.86±1.08 | 8.01±2.33 |
| ZER       | 500          | v.o | 8.09±1.29 | 11.20±2.26 | 12.67±2.33 | 13.45±1.55 | 14.51±0.49 |
| ZER       | 1000         | v.o | 5.15±0.36 | 9.39±0.32 | 12.85±0.31 | 13.49±0.26 | 14.95±0.05 |
| ZER       | 1500         | v.o | 7.18±0.39 | 14.39±1.34 | 15.05±0.39 | 15.49±0.29 | 15.95±0.85 |

Data are average ± EPM; p<0.05 in relation to control; p<0.01 in relation to control; n = 10.

to explain pain and analgesia mechanisms, with better results than those using mechanical or thermal stimuli. The model consists of two distinct phases. The first phase represents the irritant effect of formalin on the sensory C-fibers (Figure 3).

The formalin nociception model consists of the intraplantar injection of formaldehyde solution in the hind paw of the animal, which triggers intense nociception by direct stimulation of the nociceptors. Nociception caused by intraplantar injection of formalin and characterized by vigorous licking, biting and beating of the injected paw as an irritant. Most studies using this model use rodents, predominantly mice and mice. This test is characterized by two distinct stages of nociception, which seem to involve different mediators (Hunskaar et al., 1985; Hunskaar and Hole, 1987; Correa and Calixto, 1993).

The first phase of nociception begins immediately after formalin injection, extending for the first 5 min, which is believed to be due to direct chemical stimulation of the nociceptors (Hunskaar et al., 1985), predominantly of type C afferent fibers and partly of type (Heapy et al., 1987) and is associated with the release of excitatory amino acids, nitric oxide and P-substances between others.

Figure 4 shows the second stage of this model occurs between 15 and 30 min after formalin injection and is mainly related to the release of several pro-inflammatory mediators such as bradykinins, histamine, prostaglandins and serotonin, among others (Hunskaar and Hole, 1987; Correa and Calixto, 1993).
sequiterpenico ZER administered orally caused maximum antinociceptive activity in 1 h, being significant within 6 h after its administration, in relation to the second phase of the formalin induced nociception. In addition, the present data also demonstrated that the ZER compound, administered v.o, produced a significant antinociceptive effect in a dose-dependent manner in relation to both phases of formalin-induced nociception, but was more effective in relation to the second phase of this model. Thus, the results reinforce the hypothesis of the important antinociceptive and/or anti-inflammatory activity.

**Analgesic activity produced by capsaicin**

The treatment of animals with the ZER sesquiterpene (1-3%) caused significant reduction and dependent neurogenic pain induced by ear edema intraplaque injection of capsaicin. The results indicate that ZER sequiterpene at the dose of 50 mg/kg when compared with the control inhibited 70 + 5% of pain caused by capsaicin.

The results obtained in the model revealed that ZER sequiterpene, administered v.o, was able to inhibit neurogenic pain caused by capsaicin, thus opening up new perspectives for the therapeutic use of the ZER compound in these types of pains. Another important aspect in the present study was the fact that ZER showed the same efficacy and potency both administered orally and i.p, suggesting that this sesquiterpene has good availability, which makes this compound important for a future drug with analgesic activity (Figure 5).
DISCUSSION

The present study shows clearly the sesquiterpene ZER obtained from Z. zerumbet rhizomes essential oil to possess potent antinociceptive activity when administered via intraperitoneal or oral routes in the different neurogenic-induced nociception models in mice and rats. Moreover, the compound presented expressive antinociceptive activity when analyzed in the hot plate thermal sensitivity assessment nociception. Among the pain models used in this work, that of acetic acid intraperitoneally-induced abdominal writhing is relatively simple with little specificity, but easy to be observed, quick and with good sensibility to several non-steroidal antiinflammatory and analgesic drugs, as well as to drugs similar to other centrally-acting analgesics.

Through this test it was possible to demonstrate, for the first time, that ZER administered via oral route, inhibits acetic acid induced writhing up to 98% within 2 h. In an in vivo assay, an acetic acid-induced writhing response in mice was significantly reduced by treatment with zerumbone. Furthermore, zerumbone reduced paw edema and the pain response in a mono-iodoacetate (MIA)-induced rat osteoarthritis model (Chien et al., 2016). Sulamain et al. (2010) tested ZER in doses of 10, 50 and 100 mg/kg with realized inhibitions of 19.3, 40.4 and 64.8%, respectively as compared to the control (acetylsalicylic acid). It was concluded that the direct acetylsalicylic acid induction can liberate endogenous mediators, with prostaglandins E2 (PgE2) and PGF2 in peritoneal fluids. While ZER inhibits abdominal writhing through the lipoxygenase and/or cyclooxygenas mechanism by reducing the afferent nociceptors primary transduction. Experiments carried out with pharmaceutical agonists and antagonists suggest that the analgesic effect caused by substance ZER is probably related with an interaction with the opiate or glutaminergic systems without involving the L-arginine/nitric oxide pathway and sedative or muscle relaxing effects on the central and/or peripheral nervous system.

Another major aspect being analyzed in the present study is the possibility of the sesquiterpene ZER to produce analgesic effect in the supra-spinal mechanism in a nocicepuion model, which utilizes harmful thermal stimulation in the experiments using the hot plate. Doses that showed to be effective in the hot plate test, at 50°C, were similar to those that were necessary and efficient for the analgesic effect in chemical stimulation model. Furthermore, the present study demonstrates the positive control, fentanest (synthetic opiate), to be effective in also lengthening the time of latency of the animals in the thermal nociception stimulation, regardless the temperature being used. That might be a reflex either from the great difference of the stimulations, or from the integration of the responses to different temperatures (Hwang and Wilcox, 1987).

Sulamain et al. (2010) investigated the effects of ZER through the test of thermal stimilations (hot plate) following the protocol for intraperitoneal administration (10, 50 and 100 mg/kg) with a special significance in the dose of ZER having a prolonged latency in the heat stimulation. The ZER effective time in mice using the hot plate test also confirmed its anti-nociceptive activity through its action on the central mediators. Perimal et al. (2011) demonstrated that zerumbone possesses significant peripheral and central antinociceptive effects.

Given that neurogenic nociception is very complex and that there are no satisfactory therapeutic alternatives for its treatment, as of yet, these findings may open new perspectives for the development of molecules presenting therapeutic potential for treating pain (Carter, 1991; Julius and Basbaum, 2001; MacFarlane et al., 1997).

Conclusion

The findings of the present study suggest ZER in the models of acetic acid-induced nociception, thermal and
mechanical tests: hot plate and hyperalgesia to present therapeutic potential, contributing to the discovery of new drugs possessing analgesic effect, so much so as to be referred to for the control of the neurogenic nociception. Another important feature in the present study was the fact of ZER presenting the same efficacy and power, when administered in dependent dose both via oral and intraperitoneal route, suggesting this substance to present good availability.

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CONFLICT OF INTERESTS

The authors declare that they have no conflict of interest.

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