Thermotolerant glycosyl hydrolases-producing *Bacillus aerius* CMCPS1 and its saccharification efficiency on HCR-laccase (LccH)-pretreated corncob biomass

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**Abstract**

**Background:** The current production of bioethanol based on lignocellulosic biomass (LCB) highly depends on thermostable enzymes and extremophiles owing to less risk of contamination. Thermophilic bacterial cellulases are preferred over fungi due to their higher growth rate, presence of complex multi-enzymes, stability, and enhanced bioconversion efficiency. Corncob, underutilized biomass, ensures energy conservation due to high lignocellulosic and more fermentable sugar content. In the present study, the thermophilic bacterium *Bacillus aerius* CMCPS1, isolated from the thermal springs of Manikaran, Himachal Pradesh, India, was characterized in terms of its activity, stability, and hydrolytic capacity. A two-step process comprising: (i) a combined strategy of hydrodynamic cavitation reaction (HCR)-coupled enzymatic (LccH at 6.5 U) pretreatment for delignification and (ii) subsequent hydrolysis of pre-treated (HCR-LccH) corncob biomass (CCB) using a thermostable cocktail of CMCPS1 was adopted to validate the efficiency of the process. Some of the parameters studied include lignin reduction, cellulose increase, and saccharification efficiency.

**Result:** Among the five isolates obtained by in situ enrichment on various substrates, *B. aerius* CMCPS1, isolated from hot springs, exhibited the maximum hydrolytic activity of 4.11. The GH activity of the CMCPS1 strain under submerged fermentation revealed maximum filter paper activity (FPA) and endoglucanase activity of 4.36 IU mL⁻¹ and 2.98 IU mL⁻¹, respectively, at 44 h. Similarly, the isolate produced exoglucanase and β-glucosidase with an activity of 1.76 IU mL⁻¹ and 1.23 IU mL⁻¹ at 48 h, respectively. More specifically, the enzyme endo-1,4-β-D-glucanase E.C.3.2.1.4 (CMCase) produced by *B. aerius* CMCPS1 displayed wider stability to pH (3–9) and temperature (30–90 °C) than most fungal cellulases. Similarly, the activity of CMCase increased in the presence of organic solvents (118% at 30% acetone v/v). The partially purified CMCase from the culture supernatant of CMCPS1 registered 64% yield with twofold purification. The zymogram and SDS-PAGE analyses further confirmed the CMCase activity with an apparent molecular mass of 70 kDa. The presence of genes specific to cellulases, such as cellulose-binding domain CeB, confirmed the
Background

Lignocellulosic biomass (LCB)-based bioethanol pilot plants has been increasing worldwide due to concerns over climate change, energy conservation, and food security [1]. Of the several processes to produce bioethanol, the cost for saccharification of LCB is still extremely high and requires cellulases. The presently available industrial cellulases contain low levels of β-glucosidases to hydrolyze the cellobiose to glucose. In addition, lignocelluloses-based biorefineries prefer multi-beneficial cellulytic enzymes with temperature tolerance, and broader stability to pH, metals, and solvents. Such multi-beneficial cellulytic enzymes offer several advantages, such as tolerance to elevated temperatures as well as harsh operational processes, and hence reduce microbial contamination during saccharification [2]. Moreover, the increased reaction rate of enzymes reduces the viscosity of medium with a simultaneous increase in the diffusion of simple sugars from complex polysaccharides. Therefore, it is imperative to screen multi-beneficial cellulases to suit the industrial requirements [3, 4].

Multi-beneficial cellulases isolated from bacteria are preferred over fungal enzymes owing to their rapid growth rate, more extensive genetic diversity, and easy manipulation of their genetic makeup [5]. The majority of the bacterial species can produce endoglucanases to hydrolyze amorphous celluloses such as carboxymethyl cellulose (CMC). However, several of them cannot hydrolyze crystalline cellulose effectively [6]. A few Bacillus spp. possess endoglucanases with microcrystalline cellulose (Avicel)-degrading activity [7]. Furthermore, thermostolerant cellulases with β-glucosidase activity can overcome the rate-limiting steps in the saccharification process and thus increase the glucose yield [2, 8]. However, successful hydrolysis of biomass and synergistic action of cellulase depend on an optimum pretreatment process. Of the several pretreatment strategies, the biological method is promising, as there is no inhibitor formation, it requires less energy, and is eco-friendly. Several studies have suggested combined pretreatment methods due to their better delignification efficiency [9]. More recently, hydrodynamic cavitation technology coupled with an oxidative enzyme (laccase 6.5 U of Trametes versicolor) was one of the successfully employed LCB pretreatment [10]. Highly reactive radicals (H– and OH–) generated due to the cavitation effect degrade lignin moieties, although not optimal for targeting specific end products [11]. In HCR-laccase coupled process, both the biocatalyst and biomass are continuously circulated throughout the reactor. Laccase releases phenoxy radicals to remove recalcitrant fractions of lignocellulosic biomass and enhances delignification efficiency. The present study deals with multi-coperoxidase (LccH) from a hyper laccase-producing fungus Hexagonia hirta MSF2 [12] in a hydrodynamic cavitation reactor (HCR). Hence, the present study isolated multifunctional cellulases encompassing microcrystalline cellulose degradation, pH, temperature, and metal tolerance for enhanced saccharification of HCR-LccH-pretreated corn-cob biomass (CCB).

Results

Thermophilic glycosid hydrolases-producing bacteria

To develop a robust biomass conversion process that works at slightly elevated temperatures, a biotrap-based in situ enrichment was performed to isolate thermophilic bacteria. Five bacterial isolates that grow at relatively high temperature (>50 °C) were isolated from thermal springs (Manikaran [~95 °C]) of Himachal Pradesh, India, through an in situ enrichment of several lignocellulosic substrates, namely paddy straw, banana fiber, banana pseudostem, rice husk, and palm fronds. Among the five isolates screened for their cellulytic potential, CMCP51 showed the maximum hydrolytic capacity of 4.11 under the plate assay (Table 1), and was used for further studies.

Under submerged fermentation conditions, the potential isolate CMCP51 showed a filter paper activity of 3.0 IU mL$^{-1}$, endoglucanase activity of 2.0 IU mL$^{-1}$, exoglucanase activity of 1.5 IU mL$^{-1}$, and β-glucosidase activity of 1.0 IU mL$^{-1}$ at 48 h (Fig. 1), which confirmed the significant production of extracellular glycoside hydrolases (GHs) using respective substrates. The exoglucanase
and β-glucosidase activities of CMCPS1 were higher than those of B. tequilensis VCB1 and B. licheniformis KBFB3.

Phylogenetic analysis of CMCPS1
The 16S rRNA gene sequence of the isolate CMCPS1 was subjected to a bioinformatic search for their nearest matching nucleotide sequence using the NCBI-BLAST analysis tool along with the type strains retrieved from NCBI databases. The phylogeny was generated using the neighborhood joining method with MEGA 7.0 and a bootstrap value of 0.02. The results revealed that the isolate was closely related to B. aerius and formed a clade with B. aerius 2K4, showing 98% sequence similarity. The sequence of the strain was deposited in the NCBI GenBank with an accession number of MH478394 (Fig. 2).

Cellulase-specific genes and binding domains from B. aerius confirming GH activity
The strain B. aerius CMCPS1 was screened using polymerase chain reaction (PCR) for cellulase GH46 (celS and celB) and β-glucosidase (GH3)-encoding genes (bgl). The results demonstrated that the strain harbored CelB corresponding to GH46, producing a 650-bp amplicon (Fig. 3a) in comparison to B. licheniformis KBFB2, used as a positive control. There was no amplification for the celS domain. Similarly, in β-glucosidase screening, a 1500-bp amplicon was observed (Fig. 3b), suggesting that the strain confirmed the presence of glycoside hydrolase activity compared with the standard culture B. licheniformis KBFB3 (NAIMCC-B-02118).

Multi-GH complex of B. aerius CMCPS1
Results of the FP activity assay of B. aerius CMCPS1 exhibited an activity of 4.36 IU mL$^{-1}$ (Fig. 4). It was noted that as the growth (OD$_{600}$) increased, FP units showed a pronounced and steady-state increase until the early stationary phase (44 h) and gradually declined thereafter toward the late stationary phase. The strain CMCPS1 showed carboxymethyl cellulase (CMCase E.C.3.2.1.4) activity of 2.98 IU mL$^{-1}$ at 48 h, which declined gradually (Fig. 4). The CMCase production and growth pattern are similar to that of FPA. The exoglucanase activity determined using Avicel (crystalline cellulose) as a substrate recorded a maximum titer of 1.76 IU mL$^{-1}$ at 48 h of growth by CMCPS1. However, a drop in the activity was noticed after 48 h (Fig. 4). It was further noticed that unlike FPA and CMCase, the exoglucanase activity was maximum in the late stationary phase.

Cellobiase (E.C.3.2.1.21) activity or aryl β-glucosidase of the strain B. aerius CMCPS1 grown in a production medium containing cellobiose (1%) was assayed. Interestingly, the strain CMCPS1 showed an increasing trend up to 24 h, and attained a maximum titer of 1.23 IU mL$^{-1}$ at 48 h; the activity declined thereafter (Fig. 4). Hence, β-glucosidase was constitutively produced during the logarithmic phase, and its production lasted until the late stationary phase.

Characterization of CMCPS1 cellulase complex
Partially purified extracellular protein fraction of B. aerius CMCPS1 was subjected to SDS-PAGE analyses, and the results demonstrated three distinct bands of 100, 80, and 70 kDa (Fig. 5A). Native PAGE gel stained with Congo red revealed a clearing zone due to hydrolysis (Fig. 5B). Similarly, agarose well diffusion assay confirmed the CMCase activity (Fig. 5C).
Optimal assay conditions and stability of CMCase

The determination of biochemical characteristics of cellulase is of prime importance to elucidate the process parameters in industrial applications. The pH has a substantial role in cellulase activity; the maximum cellulase activity was observed at pH 7.0 in a phosphate buffer. Interestingly, partially purified cellulases retained more than 50% of their activity between pH 3.0 and 9.0 (Fig. 6a, b) after 5 h of incubation. Thermal inactivation of the enzyme is often encountered as a significant problem in most industrial processes. Industries prefer enzymes with thermophilic nature, as most of the operations in the industry work at moderately higher temperatures. The CMCase activity was maximum at 50 °C; thereafter, the activity reduced. The thermostability of CMCase from CMPS1 evaluated by pre-incubation of the enzyme extract at pH 7.0 for 1 h demonstrated its moderate thermostability between 45 and 70 °C by retaining more than 80% of the activity. However, prolonged incubation of enzyme for 5 h drastically reduced the enzyme activity, retaining 50% of CMCase activity.

The CMCase activity on several metals showed that none of the metals activated the cellulase activity, whereas a distinct inhibition was noticed. Among the metal ions tested, the relative cellulase activity was more with Zn$^{2+}$ (85%), followed by NH$_4^+$ (82%). Similarly, the activity.

Fig. 2 Phylogenetic tree for identification of the isolate CMPS1. All sequences of the associated members of the genus Bacillus sp. were aligned with the sequences of CMPS1. The tree was constructed using the 16S rRNA sequence retrieved from the database by using the neighbor-joining method. The bootstrap values were generated from 1000 replicates.
results suggested that cellulase showed the minimum relative activity and was strongly inhibited in the presence of iron (65%) and copper. Further, potassium and calcium ions exhibited moderate inhibition of cellulase activity (70 and 77%, respectively). The effect of organic solvents (30% v/v) on the relative activity of cellulase showed that the enzyme is extraordinarily stable in the presence of acetone (117.73%) and moderately stable in methanol (81.66%) and isopropanol (83.33%). It was observed that ethanol and polar solvents, such as n-hexane and toluene, reduced the cellulase activity, as evidenced by the lower residual activities of 68.6, 68.7, and 41.96%, respectively (Figs. 7 and 8).

**Pretreatment of corncob biomass using HCR coupled with LccH**

In the present study, CCB was pretreated with laccase (LccH at 6.5 U) from *H. hirta* MSF2 in an HCR for 60 min. The pretreatment helps to delignify the corncob biomass (CCB), depolymerize hemicellulose, and reduce cellulose crystallinity. Such a process will increase the porosity of the biomass, making CCB more amenable for the subsequent saccharification process. Here, the HCR-LccH pretreatment efficiently reduced the hemicellulose and lignin (6.57 and 54.1%, respectively), accompanied by an increase in cellulose (24.26%). The proximate fractions of both raw and HCR-LccH are depicted in Table 2.

The FT-IR spectral analysis of HCR-LccH-pretreated corncob samples revealed different wave numbers, functional groups, and their corresponding polymers that represented the presence of lignin, hemicelluloses, and cellulose structures. The absorption band peaks of cellulose, hemicelluloses, and lignin were shown to be stronger in the HCR-LccH-pretreated corncob samples compared to the untreated one. The performance of HCR-LccH pretreatment indicated significant changes in the structural polymers as compared to raw corncob. However, the pretreatment duration significantly influenced the lignin removal. A reduction in the peak intensity was observed at wavenumbers 950, 1031,1371,
Fig. 5 Protein profile of cellulase produced by *B. aerius* CMCP51. SDS-PAGE and zymogram analysis showing cellulytic activity. A SDS profile of cellulase produced by CMCP51 grown on culture medium supplemented with CMC; M—marker; 1—positive check *B. licheniformis* KBFB3; 2—CMCP51. B In vitro zymogram of the cellulase activity of the purified enzyme; 1a and 1b—positive check *B. licheniformis* KBFB3 in two replicates; 2a and 1b—CMCP51 in two replicates. C Gel diffusion assay of the partially purified cellulase (a) purified cellulase (b) sterile water control. The halo region indicates the cellulase activity of the *B. aerius* CMCP51.

Fig. 6 Optimal activity and stability of CMCase from *B. aerius* CMCP51. a Effect of pH on enzyme activity, b effect of pH on enzyme stability, c effect of temperature on enzyme activity, and d effect of temperature on enzyme stability.
1424, and 1634 cm\(^{-1}\) (Fig. 9). The peak at 898 cm\(^{-1}\) corresponds to the cleavage of \(\beta\)-glycosidic bonds in cellulose, whereas the bands at 1031 and 1100 cm\(^{-1}\) represent C–O–C vibrations associated with the pyranose ring and C–O–C asymmetric stretching in cellulose, respectively. Absorption spectra at 1371 and 1424 cm\(^{-1}\) represent C–H bending and symmetric CH\(_2\) bending, respectively, associated with cellulose scissoring. In addition, absorption peaks at 2895 and 3312 cm\(^{-1}\) indicate C–H stretching and –OH stretching, respectively, of intramolecular hydrogens within lignin and cellulose. The C=O stretching vibration and C–C vibration of the aromatic ring representing lignin were indicated by the absorption peaks at 1243 and 1514 cm\(^{-1}\), respectively. The acetyl groups in hemicelluloses and ester groups in lignin were represented by C=O vibrations in the band position of 1747 cm\(^{-1}\).

**Saccharification efficiency of cellulase from CMCPS1**

After pretreatment, enzymatic saccharification was performed using cellulase from *B. aerius* CMCPS1, and the reducing sugar was estimated. The highest saccharification efficiency of 55% (reducing sugars of 1.37 g g\(^{-1}\)) was obtained with 40 U cellulase per gram of dry biomass at a 5% solid loading rate. A gradual increase in the saccharification efficiency was observed, as the hydrolysis progressed up to 96 h, and after that, a sudden decline at 120 h was observed. At lower enzyme loading of 20 U g\(^{-1}\), 40% efficiency (0.99 g g\(^{-1}\) reducing sugars) was achieved in 96 h, however, doubling the enzyme load to 40 U g\(^{-1}\) increased the saccharification efficiency to 55% in 96 h. The reducing sugar yield obtained in the present study was compared with previous reports (Table 3). In contrast, the commercial enzyme from *Aspergillus niger* (0.8 U mg\(^{-1}\)) showed 8.51 to 13.07% efficiency at 72 h, which declined thereafter (Fig. 10).

**Discussion**

Thermostable cellulases offer potential merits in LCB valorization by increasing its solubility and resulting in higher reaction velocities and a paramount reduction in enzyme load [13, 14]. In addition, thermostable cellulases offer other advantages of shorter duration of hydrolysis, decreased contamination risks, increased productivity, and a more predominantly reduced cost of energy for cooling after the pretreatment process. With a prospect to search for novel thermostable GHs, it is now widely accepted that thermophilic microbes growing at a temperature of 50 to 80 °C are the nature’s reserve for hyperactive thermo-, alkali- and solvent-tolerant cellulases [15–19]. The present study isolated thermostolerant cellulases producing thermophilic bacterial strain *B. aerius* CMCPS1 obtained through
the biotrap-based enrichment of paddy straw in the hot springs (Manikaran) of India. Previously, a variety of microorganisms belonging to the genera *Bacillus*, *Geobacillus*, *Thermotoga*, *Caldibacillus*, *Acidothermus*, *Caldocellum*, and *Clostridium* were reported to produce thermostable cellulases [2]. Furthermore, an enhanced endoglucanase production by *Bacillus aerius* S52 on mixed lignocellulosic substrates was attempted [20]. However, there exist no previous reports of thermophilic *B. aerius* producing the thermostable cellulase, as the majority were reported in *B. licheniformis* and *B. subtilis* [2, 21].
Endoglucanases with carbohydrate-binding modules (CBMs) depolymerize crystalline cellulose. In the present study, \textit{B. aerius} CMCP51 exhibited a significant \textit{cellB} amplicon (650 bp), suggesting the presence of an inherent gene for endo $\beta$ 1-4 glucanase, belonging to the CBM family 46. The presence of \textit{cellB} from \textit{B. halodurans} of CBM 46 has been reported earlier [26]. However, \textit{cellS} was not found in \textit{B. aerius} CMCP51. Our previous finding [21] confirmed the presence of cellulose-binding operons of size 250 bp (\textit{celS}) domain in thermophilic bacterial strains of \textit{B. tequilensis} isolated from the hot springs of Himachal Pradesh. In addition, the $\beta$-glucosidase-specific primers generated a significant amplicon size of 1500 bp, which further confirmed the presence of multidomain GH-encoding operons. The majority of the hyperthermophilic microbes did not degrade crystalline cellulose due to the absence of a CBM [27]. However, the presence of a multidomain thermophilic cellulase of CMCP51 favors the degradation of LCB [28]. Hence, it can be concluded that \textit{B. aerius} CMCP51 possessing multidomain GH cellulase allows efficient cellulose depolymerization at high temperatures as a single source of the biocatalyst.

Partial purification of cell-free culture supernatant of CMCP51 revealed multiple banding patterns, which could be attributed to endoglucanases of cytoplasm, endoglucanases from periplasm, or aryl-$\beta$ glucosidase II [21]. In the present study, the GHs were extracellular, as evidenced by the agarose cell diffusion assay. Hence, these endoglucanases could be bound to cells and may be exoenzymes, as observed in our previous report [21] and by Li et al. [29].

The properties of cellulases, particularly enzyme deactivation and stability under adverse temperature and pH conditions, are essential for industrial applications. The broader pH stability of CMCP51 confirms its tolerance to alkali. Alkaline cellulases from alkaliphilic \textit{Bacillus} strain KSM-635, which is active at pH values higher than 8.0 under mesophilic conditions (30–37°C), has already been reported [30]. Moreover, thermostable alkaline cellulases (pH 3.0–9.0) from \textit{Bacillus licheniformis} 380, isolated from compost, have been purified and characterized [31]. In addition, acid-tolerant cellulases (pH 4.8) have been reported by several other studies on \textit{Nectria catalinensis} [32]. Hence, the results suggest the possibility of using alkali-tolerant cellulases isolated from \textit{B. aerius} CMCP51 for potential application in the detergent industry.

Thermal inactivation of GHs is often encountered in most biorefineries. The present study envisaged the thermal stability of purified cellulase over a temperature ranging from 30 to 90 °C. At 80 °C and 90 °C, the CMCase activity declined rapidly, possibly because of protein instability to higher temperatures, leading to thermal denaturation. Cellulases with an activity up
to 40 °C cannot saccharify efficiently because the enzymatic hydrolysis at ≤ 50 °C is often incomplete. Hence, higher thermostable cellulases are preferred. It has been reported that cellulases showed broader thermal stability at different incubation times, and their activity also retained 90% and 60% of the maximum activity at 60 °C and 80 °C, respectively [33]. Moreover, the CMCase of Bacillus aerius CMCP51 retained > 50% of activity beyond 80 °C. Several other thermophilic strains have been shown to produce thermostable cellulases; however, the activity was not retained at increased temperatures for more prolonged incubation [34].

In the present study, a strong inhibitory effect of cellulase activity in the presence of Fe²⁺ and Cu²⁺ ions and a more stable activity owing to acetone was observed following the earlier reports in Bacillus amyloliquefaciens DL-3 [35] and Bacillus vallismortis RG-07 [36]. The inhibition of cellulase activity by the divalent metal cations Cu²⁺ and Fe²⁺ ions might be due to (i) the prevalence of their competition with protein-associated cations, resulting in reduced metalloenzyme activity; (ii) binding with thiol groups and interaction with carboxyl or imidazole groups of amino acids [37]. However, the thermostable cellulase of Bacillus aerius CMCP51 is moderately tolerant to Zn²⁺ and Ca²⁺. In contrast, Paenibacillus sp. strain B39 showed the maximum enzyme activity in the presence of Ca²⁺, Mg²⁺, and Na⁺ ions [38].

Enzyme-catalyzed reactions can be significantly enhanced in organic solvents than in natural aqueous water-based media, which offer numerous potential applications in several industrial processes. The cellulase of Bacillus aerius CMCP51, in the presence of acetone, exhibited much higher relative activity, whereas polar solvents showed a marginal reduction. A previous report demonstrated that except for benzene, solvents propanol, ethanol, acetone, methanol, and cyclohexane increased the cellulase activity of Bacillus vallismortis RG-07 [36]. Similarly, an organic solvent-stable cellulase of Bacillus halodurans CAS1 with enhanced activity in the presence of organic solvents (25% v/v) has also been reported [39]. The stimulation of enzyme activity by organic solvents may be due to the formation of intermediary compounds of residues of non-polar hydrophobic solvents. The cellulase is thereby maintained in an open conformation, resulting in stimulated activation. We, therefore, suggest that the cellulase of thermophilic Bacillus aerius CMCP51 is highly stable in the presence of both hydrophilic and hydrophobic organic solvents at 30% v/v.

In the present study, the enzymatic pretreatment of CCB in an HCR-LccH resulted in a significant reduction in lignin (54.1%) and hemicellulose (6.57%), and increase in cellulose (24.26%), which confirms our previous finding [10]. With the pre-treated CCB, at 5% solid loading rate, we attained a saccharification efficiency of 54.59% using partially purified cellulase (40 U mL⁻¹) of CMCP51 at 96 h. Cellulase, a complex of three different enzymes, acts in synergism and a cooperative association, producing substrates for each other [17, 40]. The decrease in the sugar yield at low enzyme loading beyond the maximum solid loading concentration might be attributed to the presence of inhibitors, as observed earlier [41]. The accumulated cellobextrin and cellobiose with a lower degree of polymerization might inhibit hydrolysis. A saccharification yield of 46% was also obtained previously using cellulase from Trichoderma citrinoviride using perennial grass Artemisia annua under pretreated conditions [42]. Similarly, the one-pot consolidated bioprocessing of ionic liquid-pretreated pine needles by Bacillus subtilis (SV1) achieved a maximum saccharification efficiency of 65.9% [43]. The difference in sugar yield might be caused by the pretreatment method or nature of biomass used. In the present study, HCR-LccH pretreatment resulted in a higher degree of delignification. The cellulase of CMCP51 has more accessibility to cellulose, at 5% solid loading rate, and an enzyme loading rate of FP units 40 U g⁻¹ recovered higher sugar from CCB compared to the previous reports [10, 52, 53]. Yet this is the first report on thermo-alkali-tolerant multi-domain GH of Bacillus aerius CMCP51 and its saccharification efficiency in HCR-LccH-pretreated biomass.

Conclusions
Thermophilic cellulolytic bacterial strain CMCP51, isolated from an enriched biomass sample at the hot springs of Himachal Pradesh, was identified as Bacillus aerius CMCP51. In general, the stability of the partially purified cellulase of CMCP51 over a range of pH, temperature, metal ions, and organic solvents offers a wide scope for its applications in biorefineries to produce sugars and concomitant fermentation products. The activity over a wide range of pH (3–9) and temperature (30–90 °C) clearly indicates the thermoalkaline nature of cellulase from CMCP51. The saccharification efficiency of cellulase from CMCP51 was 55% on HCR-LccH-pretreated corn cob at the maximum enzyme and substrate loading rate of 40 U g⁻¹ and 5%, respectively. There exists further scope of enhancing the saccharification efficiency by staggered enzyme loading, making an enzyme cocktail, and optimizing the conditions. Moreover, the strain is unique with respect to cellulase activation in the presence of several hydrophobic solvents, which make the strain a cost-efficient resource for thermo-alkali and solvent-tolerant cellulase production and subsequent bioconversion of LCB into fuels and platform chemicals.
Methods

In situ enrichment, isolation, screening, and identification of thermophilic cellulase-producing bacteria

An in situ enrichment with several cellulose-rich natural substrates was performed at the mouth of hot springs in Manikaran (~95 °C), Himachal Pradesh, India (32.0268° N, 77.3511° E). Perforated tubes with the conical bottom (15 mL) containing 200 to 300 mg of substrates (paddy straw, banana fiber, banana pseudostem, rice husk, palm fronds) placed in the spring for 2 weeks were used for isolating the microbes. Isolation, screening, and molecular characterization of thermophilic bacteria were performed as described previously [21]. Thermophilic cellulase-producing bacteria were isolated by dilution plate on a basal medium supplemented with 1% CMC. The plates incubated at 50 °C for 24 to 48 h were flooded with 0.1% Congo red, followed by destaining with 1 M NaCl [44]. The hydrolytic capacity of the bacterial isolates was calculated as given below:

\[ \text{Hydrolytic capacity} = \frac{\text{diameter of the clearing zone (cm)}}{\text{diameter of the colony (cm)}}. \]

The cellulolytic activity was compared with that of standard cultures B. tequilensis VCB1 (NAIMCC–B-02117) and B. licheniformis KBFB3 (NAIMCC–B-02118), obtained from the National Bureau of Agriculturally Important Microbes (NBAIM, ICAR-India).

Cellulase production by B. aerius CMCPS1 under submerged fermentation

Growth and cellulase production by CMCPS1

A single colony inoculated into 5 mL of Luria–Bertani (LB) broth was incubated overnight at 50 °C and further transferred into 100 mL of basal media amended with 1% CMC as the substrate [45]. The growth was monitored periodically for 78 h at 50 °C, and the OD₆₀₀ was measured at every 4 h interval. The growth rate was determined using the commercially available software GraphPad Prism 5.0.

Preparation of crude enzyme extract

The overnight-grown culture (OD₆₀₀ of 0.6) was inoculated into 50 mL of BPS-X production medium as described previously [21] supplemented with 1% CMC for cellulase production [46]. The flasks were incubated at 50 °C in an orbital shaker at 110 rpm for 168 h. The supernatant collected at every 24-h interval was centrifuged at 10,000 rpm for 10 min at 4 °C, and the cell-free culture supernatant served as the source enzyme for the following assays.

Partial purification of cell-free crude cellulase

The bacterial culture was harvested by centrifugation at 10,000 rpm for 20 min at 4 °C. The supernatant saturated with 80% ammonium sulfate was frozen and incubated for 12 h. The incubated sample was centrifuged at 10,000 rpm for 20 min at 4 °C to remove salts. The pellet resuspended in 100 mM phosphate buffer, pH 7.0, was assayed for enzyme activity and determined for protein content.

Enzyme activity measurement

Assays for total cellulases (FPA), CMCase, Avicelase, and cellobiose from cell-free crude extract were performed according to standard protocols. Respective substrates, i.e., filter paper, CMC, Avicel, and cellobiose, were used to monitor the enzyme activity as described [21]. One unit of enzymatic activity is defined as the amount of enzyme that releases 1 µmol of reducing sugars (measured as glucose) per mL per min.

Screening for cellulase genes

Thermophilic CMCPS1 strain was further screened by PCR for cellulase-encoding genes using gene-specific primers (Additional file 1: Table S1), and the amplicons were resolved in 1% agarose gel.

CMCase activity on an agarose plate

CMC (0.1%) containing agarose plates were prepared with wells. Approximately 20 µL of the partially purified enzyme was added to the wells drilled on CMC agarose plates and incubated overnight at 50 °C for 12 h. The dishes were washed with distilled water, and staining (0.1% Congo red solution) and destaining (1 M NaCl) were performed as mentioned above to detect the clearing zones around the wells due to the hydrolytic activity [47].

Sodium dodecyl sulfate–polyacrylamide gel electrophoresis analyses

Sodium dodecyl sulfate–polyacrylamide gel electrophoresis (SDS-PAGE) was performed with 12% resolving and 4% stacking gels, according to Laemmli’s method [48]. After electrophoresis, protein bands were detected as described previously [21].

Native PAGE and zymogram

The partially purified culture supernatants were run on 12% non-denaturing PAGE. The gel was placed in the agarose plate containing 0.1% CMC and incubated for 1 h at 50 °C [47].
**Protein determination**

The protein concentration was determined [49] by mixing 1 mL of Bradford's reagent (BioRad) with 50 µL of the sample and reading in a multimode microtiter plate reader (SpectraMax@i3x) at 595 nm using bovine serum albumin as the standard. The specific activity of CMCase was calculated and expressed in terms of IU per mg of protein.

**Optimization of assay conditions**

**Effect of temperature, pH, solvent, and metals on enzyme activity**

The optimum temperature, pH, solvent, and metal ion concentration for the enzyme activity were monitored. The thermal stability was monitored by incubating the enzyme with 1% CMC in 100 mM phosphate buffer under different temperatures ranging from 30 to 95 °C. The optimum pH of the purified cellulase was determined by pre-incubating the mixture of the purified enzyme and 1% (w/v) CMC in the presence of appropriate buffers; 100 mM acetate buffer (pH 3.0), 100 mM phosphate buffer (pH 7.0), and 100 mM Tris–HCl buffer (pH 8.5), and ammonia buffer (pH 9.5). The reaction mixtures were pre-incubated at 50 °C for 60 min and assayed for the activity. The pH stability was determined by incubating the purified enzyme in respective buffers having different pH ranges from 3.0 to 10.0 at 50 °C for 5 h. The residual activity of each sample for hydrolysis of CMC was subsequently estimated under assay conditions, as described above.

The effect of various metal ions (5 mM) on the enzyme activity was determined using NH₄Cl, KCl, CaCl₂, ZnSO₄, CuSO₄, and FeSO₄. The enzyme was incubated with different metals at 50 °C for 1 h, and the cellulase activity was determined under standard assay conditions.

For the solvent tolerance test, the partially purified enzyme was incubated with 30% (v/v) of different organic solvents (methanol, ethanol, acetone, isopropanol, n-hexane, and toluene) in screw-capped tubes and incubated at 50 °C for 1 h. The residual activity of each sample for hydrolysis of CMC was subsequently determined under standard assay conditions.

The optimum temperature, pH, solvent, and metal ion concentration for the enzyme activity were monitored. The thermal stability was monitored by incubating the enzyme with 1% CMC in 100 mM phosphate buffer under different temperatures ranging from 30 to 95 °C. The optimum pH of the purified cellulase was determined by pre-incubating the mixture of the purified enzyme and 1% (w/v) CMC in the presence of appropriate buffers; 100 mM acetate buffer (pH 3.0), 100 mM phosphate buffer (pH 7.0), and 100 mM Tris–HCl buffer (pH 8.5), and ammonia buffer (pH 9.5). The reaction mixtures were pre-incubated at 50 °C for 60 min and assayed for the activity. The pH stability was determined by incubating the purified enzyme in respective buffers having different pH ranges from 3.0 to 10.0 at 50 °C for 5 h. The residual activity of each sample for hydrolysis of CMC was subsequently estimated under assay conditions, as described above.

The effect of various metal ions (5 mM) on the enzyme activity was determined using NH₄Cl, KCl, CaCl₂, ZnSO₄, CuSO₄, and FeSO₄. The enzyme was incubated with different metals at 50 °C for 1 h, and the cellulase activity was determined under standard assay conditions.

For the solvent tolerance test, the partially purified enzyme was incubated with 30% (v/v) of different organic solvents (methanol, ethanol, acetone, isopropanol, n-hexane, and toluene) in screw-capped tubes and incubated at 50 °C for 1 h. The residual activity of each sample for hydrolysis of CMC was subsequently determined under standard assay conditions.

**Pretreatment of corncob biomass by HCR-LccH**

The CCB was dried to remove the moisture, and the size was reduced in a sequence shredder, pin mill, and grinder. The powdered biomass was sieved using ASTM 70 sieve as per the ASTM E11-13 procedure. The size of the biomass particles used for the pretreatment was less than 212 µm. Pretreatment of CCB was performed in an HCR at a 5% solid loading rate with laccase from *H. hirta* LccH [12] at 6.5 U g⁻¹ for 60 min [10]. The compositional analyses of the feedstock CCB before and after pretreatment was determined as per the standard protocols of NREL, 2004 [50].

**FT-IR analysis**

The FT-IR spectra of HCR-LccH-pretreated CCB were obtained on an FT-IR instrument (FT-IR 6800 JASCO, Japan) using KBr discs containing 1% finely ground samples. The absorbance spectra were recorded with a spectral resolution of 4 cm⁻¹ and 64 scans per sample between wavenumbers 4000 and 400 cm⁻¹.

**Enzymatic saccharification of HCR-LccH-pretreated CCB**

Saccharification was performed by suspending the pretreated biomass in 0.1 M sodium citrate buffer (pH 5.0) in capped polycarbonate flasks, and the substrate concentration was maintained at 10% (w/v). The crude CMCase was added in different doses (20, 30, and 40 U g⁻¹). Hydrolysis was performed at 50 °C for 120 h at 120 rpm in a water bath shaker and slightly modified from the method described previously [51]. Samples drawn at every 24 h were centrifuged, and the supernatants were analyzed for reducing sugars at 540 nm [39]. Enzymatically pre-treated CCB with commercial cellulase of *A. niger* (Sigma cat no. C1184) served as the control.

**Supplementary information**

**Additional file 1: Table S1.** Primers used for screening GHs.

**Abbreviations**

CMCase: Bacterial isolate obtained by in situ enrichment of lignocellulose substrates from hot springs of Himachal Pradesh; LccH: Crude laccase from *Hexagonia hirta* MSF2; FPA: Filter paper assay; HCR: Hydrodynamic cavity reactor; LCB: Lignocellulosic biomass; SDS-PAGE: Sodium dodecyl sulfate–polyacrylamide gel electrophoresis; GHs: Glycosyl hydrolases; CBM: Carbohydrate-binding module.

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**Authors’ contributions**

SU received the research grant and conceptualized the idea. GM, MVR, and ST conducted the experiments. SU, ST, and IM provided guidance to conduct the experiments. GM, MVR, IM, and ST wrote the original draft. SU, IM, and ST wrote, reviewed, and edited the manuscript. All authors read and approved the final manuscript.

**Data availability**

All data of this manuscript are included in the manuscript. No separate external data source is required. Any information required will be provided by communicating with the corresponding author via the official mail: usiva@tnau.ac.in.
Ethics approval and consent to participate
No specific permissions were required for the described field studies because no human or animal subjects were involved in this research.

Consent for publication
All authors agreed to publish the content.

Competing interests
The authors declare that they have no competing interests.

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