Phytochemical analysis, antioxidant and anti-inflammatory potential of FERETIA APODANTHERA root bark extracts

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Abstract

Background: Inflammation has been implicated in many disorders, including cancer and available therapies elicit adverse effects. Plants of the family Rubiaceae have shown potency against inflammation. The anti-inflammatory and anti-oxidant potential of Feretia apodanthera was investigated in this study to evaluate its effectiveness.

Methods: The phytochemical, antioxidant and anti-inflammatory potential of root bark (n-Hexane, diethyl ether, ethanol and aqueous) extracts of Feretia apodanthera was investigated in this study. The extracts were subjected to various chemical tests for phytochemical constituents; their antioxidant activity was determined using in-vitro DPPH radical scavenging activity assay and their anti-inflammatory activity was determined using carrageenan induced paw oedema model. FTIR and GCMS analysis was done to determine the compounds present.

Results: Phytochemical screening of extracts revealed the presence of unsaturated steroids, triterpenes, cardiac glycosides, tannins, saponin and alkaloids. Vitamin C had a median inhibitory concentration (IC50) of 0.038 mg/ml which was lower than IC50 of all the extracts. Of all the extracts, ethanol extract had the lowest IC50 (0.044 mg/ml) which is comparable to vitamin C. Anti-inflammatory studies showed that the inflammation inhibition potential of 400 mg/kg body weight of all the extracts was significantly lower (p < 0.05) than the standard ketoprofen (50 mg/kg) at the first three hours but significantly higher (p < 0.05) at the fourth hour. At the fifth hour, the inflammation inhibition potential of diethyl ether, ethanol and aqueous extracts were significantly higher (p < 0.05) than that of the standard. FTIR analysis showed the presence of ketones, amines, alkenes and carboxylic groups. GCMS analysis revealed compounds that are potential anti-inflammatory agents.

Conclusion: This study revealed that extracts of Feretia apodanthera possess anti-inflammatory effects against right hind paw oedema of albino rats and can act as an effective antioxidant.

Keywords: Phytochemical, Antioxidant, Inflammation, Feretia apodanthera, Extracts
Background
Plants have in one way or the other provided the basic need of mankind – food, shelter, clothing, protection from disease causing agents and treatment of various infections and illnesses since early days of human history [1]. Medicinal plants comprise of phytochemicals that improves the physiological balance of human beings and the knowledge of these healing properties has been passed down through generations. Traditional medicine is used by many African countries like Cameroon, Mali, Nigeria and Zambia to meet their health care needs [2, 3]. A vast range of medicinal plants around the globe has not yet been investigated to ascertain the claims made by traditional folks about their usefulness in treating diseases. Therefore, future research can be focused on medicinal plants.

Inflammation is usually a body response to tissue damage and to a number of systemic malfunctions including asthma, atherosclerosis, arthritis, physical injury and infection amongst many others [4]. Pro-inflammatory cytokines such as interleukin (IL)–1β, tumour necrosis factor (TNF)–α, and vascular endothelial growth factor (VEGF) mediate key points in inflammation [5]. Inflammation may be caused by foreign bodies, physical injury, allergens, radiation, stress, trauma, frostbite and alcohol amongst other things. Inflammation serves to isolate and eliminate infectious agents, induce repair and give protective response. Unfortunately, inflammation can lead to pain and discomfort that can persist for a long period of time [6]. Inflammation is manifested as pain, redness, loss of function of cells, swelling and heat. Based on its duration, inflammation is classified into acute and chronic inflammation.

_Feretia apodanthera_ Del, is from the family Rubiaceae, a bush delicious shrub with winding or twisting branches, growing from 2 to 6 m tall [7], the leaf plate are elliptic to ovate in shape with sometimes apiculate apex and rounded or cuneate base. The fruits are about 3-7 mm in diameter. It is commonly known as kuru-kuru by Hausa/Fulani - Northern Nigeria [8]. Traditionally, the root decortion of _F. apodanthera_ is used to enhance erectile function of the penis and to treat stomach upset by the Hausa tribe of Kano State of Nigeria [8, 9].

Extracts from different part of the plant including the back, root and leaves have been used to treat renal and urinary infections, stomach ache, nausea, syphilis, infective wounds and other disease conditions [10, 11]. _Feretia apodanthera_ is also used to enhance erectile function of the penis in some parts of Nigeria [8]. Mental conditions had been calmed using its extract. Taiwe et al. [12] reported that the extract of _F. apodanthera_ was effective against cognitive errors, oxidative stress and seizures and in repairing memory impairment. Harman and tetrahydroharman isolated from methanol extract of _F. apodanthera_ showed antimalarial activities and also a low cytotoxicity [13]. Recently, _Feretia apodanthera_ extracts have also been shown to have high antioxidant activity similar to quercetin and relatively high flavonoid content [11]. The presence of these metabolites may be responsible for the therapeutic effect exhibited by this plant. Investigations with extract of _Feretia apodanthera_ in rats also showed a decreased the activity of nuclear factor kappa β and nitric oxide which have been implicated in inflammation [14, 15].

The above mentioned factors have thus generated research interest in carrying out this study, thus this work is design to evaluate in-vitro antioxidant capacity and anti-inflammatory effect of different extracts of _Feretia apodanthera_ against oedema in right hind paw of albino rats.

Materials
Plant material collection and identification
Fresh roots of _Feretia apodanthera_ Del were gotten from its natural habitat at a local farm at Magami village, Gusau Local Government Area, Zamfara State, Nigeria in October, 2014. It was identified and authenticated at the Department of Biological Sciences (herbarium unit), Faculty of Science, Ahmadu Bello University, Zaria, Nigeria where a voucher number (930) was deposited for the specimen.

Experimental animals
In this study, twenty five albino wistar rats of both sexes with weight ranging from 100 to 200 grammes were used. They were obtained from the National Institute for Trypanosomiasis Research animal house, Kaduna and housed based on the necessary environmental and nutritional conditions throughout the experiment. They were kept in polypropylene cages with paddy house bedding under standard laboratory condition for 14 days before the experiment was performed in order to acclimatize them. This study was permitted by Scientific and Biochemical Ethics Committee of the Faculty of Science, Ahmadu Bello University, Zaria, Kaduna State, Nigeria. The animals were provided with water ad libitum and laboratory chow. Best procedures were followed to minimize the animal used and reduce animal suffering.

Method
Preparation and extraction of plant material
The root bark was washed and then peeled with a clean knife by scraping. The scrapes were shade dried for two weeks at room temperature. The root back was further
pulverized to powder form using a miller at Institute for Agricultural Research (IAR), Zaria. The powder was kept in a cool dry place for further use.

A modified method as described by James et al. [16] was used in this part of the study. The powdered sample (50 g each) was exhaustively extracted for 48 h by cold maceration extraction method using 500 ml of distilled water, ethanol, diethyl ether and n-hexane respectively. The mixture was passed through a mesh sieve (1 mm), filtered using Whatman filter paper no. 1, and then concentrated in a water bath at 45 °C. The extracts were kept in a deep freeze in a sample bottle awaiting use.

Qualitative phytochemical screening of the extracts
Qualitative phytochemical screening was done on the four extracts of Feretia apodanthera using standard procedures described by Evans [17] to determine the phytochemicals present. The extracts (5 mg) were dissolved in 50 ml of the respective solvents used for their extraction. The solution was made ready for qualitative phytochemical analysis for carbohydrates, free reducing sugar, anthracene derivatives, cardiac glycosides, saponin, tannin, flavonoid, alkaloids, unsaturated sterols and triterpenes.

In-vitro DPPH free radical scavenging antioxidant activity
The anti-oxidant activities of plant extracts was assessed using their ability to scavenge the activity of the free radicals of the stable 1,1-diphenyl-2-picrylhydrazyl (DPPH). The method of Chan et al. [18] was employed.

Principle
The 1,1-diphenyl-2-picrylhydrazyl (DPPH) is an oxidant having in its structure an odd electron. Its purple colour is reduced to yellow coloured diphenylpicrylhydrazine when it in contact with an antioxidant that can release a hydrogen atom or electron to it. The change in colour was measured at 520 nm using a UV/Visible light Spectrophotometer.

Procedure
DPPH solution was made by adding 6 mg of DPPH in 100 ml of methanol and allowing it to dissolve. About 2 ml of DPPH solution (0.1 mM) was added to 1 ml of various concentrations of the extracts (0.020, 0.040, 0.06, 0.080, 0.100 mg/ml). A mixture of methanol and DPPH was used as control. All mixtures were vigorously shaken and made to stand in a dark place for 30 min. After this, solution absorbance was measured at 520 nm using a spectrophotometer. The experiments were performed in triplicates and the percentage scavenging activity of the extracts on DPPH radical was calculated on the basis of the formula below:

\[ \text{Scavenging activity} = \left( \frac{1 - \text{Absorbance of the sample}}{\text{Absorbance of control}} \right) \times 100 \]

IC\textsubscript{50} values were used to express the ability of the extracts to scavenge DPPH. The term “IC\textsubscript{50}” which connoted the concentration of the extract needed to scavenge 50% of DPPH radical, was calculated using the graph of scavenging activity plotted against sample concentration using Microsoft excel software.

In-vivo anti-inflammatory activity determination of the extracts
A preliminary in-vivo anti-inflammatory test on carrageenan induced hind paw inflammation was carried out on the crude extracts to determine the most potent of them using a modified method of Katak et al. [19]. The experimental animals was fasted for overnight prior to induction of edema, water was however available ad libitum.

Records of the weight of the rats were taken and the rats randomly separated into 5 cages representing 5 groups consisting of 3 rats each by matching rat with the highest weight with that of lowest weight and vice versa. The animals were fasted overnight prior to this investigation. Sub-plantar injection of 0.1 ml of 1% carrageenan in distilled water into the right hind paws of the animals was used to induce acute inflammation paw edema [20]. Water, ethanol, diethyl ether and n-hexane extracts were administered orally to each group just before carrageenan injection. Paw edema (expressed as an increase in paw volume) was measured using a digital caliper. Animals were deprived of water during this experimental period to ensure uniform hydration and reduce variability in edematous response. The animal grouping was as follows:

**Group A:** Inflammation Control. Rats were induced with 0.1 ml of 1% carrageenan in distilled water. IC

**Group B:** Rats were induced with 0.1 ml of 1% carrageenan and treated with 400 mg/kg of the n-hexane extract of Feretia apodanthera. IH

**Group C:** Rats were induced with 0.1 ml of 1% carrageenan and treated with 400 mg/kg of the diethyl ether extract of Feretia apodanthera. ID

**Group D:** Rats were induced with 0.1 ml of 1% carrageenan and treated with 400 mg/kg of the ethanol extract of Feretia apodanthera. IE

**Group E:** Rats were induced with 0.1 ml of 1% carrageenan and treated with 400 mg/kg of the aqueous extract of Feretia apodanthera. IA
Group F: Rats were induced with 0.1 ml of 1% carrageenan and treated with 50 mg/kg of the standard Ketoprofen. IS

The volume of right hind paw was measured at 1st, 2nd, 3rd, 4th and 5th hours after carrageenan injection. Increase in paw thickness and percent inhibition was calculated as below;

\[
\text{Increase in Paw Volume} = V_t - V_0
\]

Where;
\( V_t \) = Paw Volume at Time \( t \).
\( V_0 \) = Paw Volume at Time 0

\[
\text{% Paw volume Inhibition} = \frac{V_c - V_t}{V_c} \times 100
\]

Where;
\( V_c \) = Paw Volume Increase in Control Animals.
\( V_t \) = Paw Volume Increase in Treated Animals.

Further analysis was carried out on the most potent fraction.

Fourier transform infra-red spectroscopy analysis
The FTIR spectra of the fractions with the highest anti-inflammatory activity were carried out using FTIR-8400S spectrophotometer (Shidmazu model), available at the National Research Institute for Chemical Technology (NARICT) Laboratory Zaria. The paste from the ethanol and fractions were used in a form of a thin film, held in between two potassium bromide discs. The liquid paste was dropped on each disc and they spread into a thin film. The disc was then mounted in the FTIR spectrometer in the range \( 1.20 \times 10^{13} - 1.20 \times 10^{14} \text{ Hz} \) within electromagnetic spectrum of infrared section. Absorption is written in terms of wavenumbers (units cm\(^{-1}\)).

GCMS analysis of sample
Components of the sample were analyzed using a GC (Agilent 7890B) equipped with a HP-5 ms ultra inert column and coupled to a Mass spectrometer (Agilent 5977A). The sample was dissolved in methanol prior to analysis.

Statistical analysis
The results were expressed as mean ± SD. Analysis of variance (ANOVA) was used to analyse data by the SPSS program (version 20.0 SPSS Inc. Chicago, IL, USA). Duncan Multiple Range Test was used to determine if the difference between the action of the different extracts across the animal groups were significant or not. \( P \) values less than 0.05 was considered significant (\( p < 0.05 \)). The components were finally determined by matching the peaks’ mass spectra with that of the National Institute of Science and Technology (NIST) libraries mass spectral database at the Department of Analytical Chemistry, Ahmadu Bello University, Zaria.

Result
The percentage yield of extracts of Feretia apodanthera
The percentage yield of n-hexane, diethyl ether, ethanol and aqueous extracts of Feretia apodanthera is presented in Table 1. Aqueous extract had the highest yield (14.94%), ethanol extract had a yield of 9.20%, followed by diethyl ether extract (6.00%), while n-hexane extract has the lowest yield (4.54%). This showed that the percentage yield of crude F. apodanthera root bark extracts increased as polarity of the solvent of extraction used increased.

The qualitative determination of phytochemicals in n-hexane, diethyl ether, ethanol and aqueous extracts of Feretia apodanthera
The qualitative determination of phytochemicals present in the n-hexane, diethyl ether, ethanol and aqueous extracts of Feretia apodanthera is presented in Table 2. Result showed carbohydrates, unsaturated sterols, reducing sugars, and triterpenes were present in all the four extracts. Tannins and flavonoids were present in diethyl ether, ethanol and aqueous extracts; reducing sugars and saponin were present in ethanol and aqueous extracts only. Ethanol extract contained a trace amount of alkaloids. Free anthraquinone were absent in all four extracts.

The free radical scavenging ability of extracts on DPPH
A plot of percentage inhibition of free radicals versus extracts concentration depicts the antioxidant activity of the extracts of Feretia apodanthera based on their scavenging ability on DPPH which is a stable purple coloured radical and in the process reducing it into yellow coloured diphenylhydrazine. Antioxidant potential is inversely proportional to inhibitory concentration (IC\(_{50}\)) value which was calculated from the

| Table 1 Percentage Yield of n-Hexane, Diethyl Ether, Ethanol and Aqueous Extracts of Feretia apodanthera Root Bark |
| Extract          | Amount recovered (g/50 g) | Percentage yield (%) |
|------------------|---------------------------|----------------------|
| n-Hexane         | 2.72\(^a\)                | 4.54                 |
| Diethyl ether    | 3.00\(^a\)                | 6.00                 |
| Ethanol          | 4.60\(^a\)                | 9.20                 |
| Aqueous          | 7.47\(^c\)                | 14.94                |
| S.E.M            | 1.09                      |                      |
| P value          |Group\(_{\text{F}}\)                   | 0.05                 |

Values with different superscript down the column are significantly different (\( p < 0.05 \)). S.E.M Standard error of mean
The results of inhibition study are presented in Fig. 1 and Table 3. The results in Table 3 presents the 50% Inhibitory Concentration (IC$_{50}$) values, the equation formular and DPPH radical scavenging activities at 1.00 mg/ml of n-hexane, diethyl ether, ethanol and aqueous extracts of *Feretia apodanthera*. The result shows that ethanol extract has the lowest IC$_{50}$ value (0.053 mg/ml), followed by aqueous (0.064 mg/ml), then n-hexane extract (0.7499 mg/ml) and diethyl ether extract had the highest value of IC$_{50}$ value (1.296 mg/ml) compared with vitamin C which had an IC$_{50}$ of 0.048 mg/ml. The percentage free radical scavenging activity of the extract also showed that of all the extracts, ethanol extract had the highest percentage scavenging activity at all concentrations though it was significantly ($p<0.05$) lower than the scavenging activity of vitamin C at the third, fourth and fifth hours respectively.

**Table 2** The Qualitative Determination of Phytochemicals in n-Hexane, Diethyl Ether, Ethanol and Aqueous Extracts of *Feretia apodanthera*

| Phytochemical          | Method       | N-hexane | Diethyl ether | Ethanol | Aqueous |
|------------------------|--------------|----------|---------------|---------|---------|
| Carbohydrates          | Molish       | –        | –             | +       | +       |
| Reducing Sugars        | Fehling      | –        | –             | +       | +       |
| Free Anthraquinone     | Bontrager    | –        | –             | –       | –       |
| Unsaturated Steroids   | Salkowski    | +        | +             | +       | +       |
| Triterpenes            | Liebermann-Buchard | +     | +             | +       | +       |
| Cardiac Glycosides     | Keller-Killani | –     | +             | +       | +       |
| Saponin                | Frothing     | –        | –             | +       | +       |
| Tannins                | Ferric chloride | –     | +             | +       | +       |
| Flavonoids             | Shinoda      | –        | +             | +       | +       |
| Alkaloids              | Dragendoff   | –        | –             | +       | –       |

+ = Present, − = Absent

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**Anti-inflammation studies of Feretia Apodanthera extracts**

The effect of the aqueous, ethanol, diethyl ether, n-Hexane extract of the root bark of *F. apodanthera* and the standard (ketoprofen) on carrageenan induced inflammation is presented in Table 4. The anti-inflammatory effects of the n-hexane, diethyl ether, ethanol and aqueous extracts of *F. apodanthera* was significantly ($p>0.05$) lower than ketoprofen on carrageenan induced inflammation for the first, second and third hours. At the fourth and fifth hours, all the extract s exhibited a significantly ($p<0.05$) higher anti-inflammatory potential than ketoprofen except for the n-hexane extract which showed no significant difference from the standard at the fifth hour only. Of all the extracts, ethanol extract had the highest inhibition of 93.35%.

**FTIR spectra for ethanol extract of *F. apodanthera***

Figure 2 shows the FTIR functional group spectra while Table 5 presents the summary of functional groups identified. Halide, alkyl, alkene, ketone, carboxylic acid were identified in the fraction.

**Discussion**

Many plant antioxidant potentials are related to their therapeutic potentials [21, 22]. The qualitative phytochemical screening of the extracts of *Feretia apodanthera* revealed the presence of saponins,
triterpenes, tannins, flavonoids, cardiac glycosides and steroids which might be responsible for the obvious anti-inflammatory activities of the extracts of the plant. This is in line with the report of Ahmadiani et al. [23] who stated that flavonoids as well as tannins possess anti-inflammatory effects. Flavonoids such as Quercetin are revealed to be useful in acute inflammation [24, 25]. Some flavonoids have a significant inhibitory potential against a wide array of enzymes such as phosphodiesterases, phospholipase A2, protein tyrosine kinases, protein kinase C, and others [26]. Manthey et al. [27] also reported that a number of flavonoids act by inhibiting key enzymes that are central in synthesizing prostaglandins processes. Triterpenoids may perform its anti-inflammatory action by reducing the cells that expresses inducible nitric acid synthase (iNOS), eg lupeol [28] or by inhibiting the production of nitric oxide by decreasing iNOS expression [29]. This shows that the high potential of these phytochemical compounds to inhibit inflammation processes is due to a decline in pro-inflammatory cytokines and iNOS production [30, 31]. Alkaloids is said to reduce the intensity of oedema caused by carrageenan by inhibiting vascular permeability induced by histamine [32].

The significant (p < 0.05) anti-inflammatory effect of the diethyl ether, aqueous and ethanol extracts may be due to the presence of flavonoids, triterpenes and tannins while the reduced anti-inflammatory effect of n-hexane is probably due to the absence of tannin, saponin and flavonoids. Ethanol extract showed the highest anti-inflammation potential, this may be due to a higher intensity of triterpenes and flavonoids, along with the presence of saponin, tannin, cardiac glycosides, steroids and alkaloids. This is in agreement with the work of Han and Bakovic [31] which suggests that triterpenoids are biologically active in producing anti-inflammatory effects and with the research done by Hamaleinen et al. [33] that showed that flavonoids inhibits inflammation by inhibiting signal transducer and activator of transcription 1 (STAT-1) and nuclear factor kappa beta (NF-kB) activations.

One of the primary methods for evaluating radical scavenging activity is by evaluating the ability of the antioxidants present in an extract to reduce purple coloured 1,2-diphenyl 2-picryl hydrazyl (DPPH) radical to yellow coloured diphenylpicrylhydrazine [31, 34]. A higher DPPH radical-scavenging activity is associated with a lower IC50 value [32]. Therefore, the ethanol extract had the highest DPPH reducing activity based on its relatively low IC50 values which was comparable with vitamin C as no significant (p < 0.05) difference was observed between their IC50 (Table 3).

### Table 4 The Percentage Inhibition of the Aqueous, Ethanol, Diethyl Ether and n-Hexane Extract of the Root Bark of *Feretia Apodanthera* on Carrageenan Induced Inflammation

| Extract              | 1st Hour (%) | 2nd Hour (%) | 3rd Hour (%) | 4th Hour (%) | 5th Hour (%) |
|----------------------|--------------|--------------|--------------|--------------|--------------|
| n-Hexane (400 mg/kg) | 12.47 ± 9.99a| 24.11 ± 15.53a| 18.21 ± 6.97a| 77.83 ± 21.20a| 30.28 ± 25.74a|
| Diethyl ether (400 mg/kg) | 23.13 ± 7.15a| 33.18 ± 9.00a | 20.65 ± 13.11a| 87.02 ± 8.83b | 87.82 ± 8.21b |
| Ethanol (400 mg/kg)   | 16.85 ± 13.56a| 33.65 ± 20.59a| 30.43 ± 15.86a| 89.73 ± 4.52b | 93.35 ± 4.88b |
| Aqueous (400 mg/kg)   | 23.45 ± 11.18a| 19.02 ± 10.14a| 14.48 ± 14.13a| 82.85 ± 11.93b| 88.12 ± 7.80b |
| Standard (50 mg/kg)   | 45.90 ± 11.17b| 69.94 ± 12.49b| 72.55 ± 18.21b| 45.83 ± 17.02a| 48.23 ± 23.78a|

Values are written as means ± SD n = 3 replicate determinations values with different superscript down the column are significantly different (p < 0.05). Values with similar superscripts are not significantly different.
A positive result by the aqueous and ethanol extracts in this test indicates that they contain antioxidants that can scavenge free radicals [35]. Ethanol extract was followed by the aqueous extract, the n-hexane extract and the diethyl ether extract showed the least DPPH reducing activity due to its relatively high IC50 values. Previous studies on the DPPH radical scavenging of the aqueous and aqueous-acetone extract of F. apodanthera by Coulibaly et al. [11] showed lower IC50 values than all the extracts. This may be due to the difference in sample or extraction variation.

Carrageenan induced rat paw oedema is a well-established animal model for evaluating the anti-oedematous effect of drugs or compounds [36]. From the data presented on the anti-inflammation activity of the crude extracts, oedema was mildly inhibited by 400 mg/kg of the aqueous, ethanol, diethyl ether and n-hexane extracts from the first three hours (varying from 14.47 to 33.65%). This is in line with the report by Stark et al. [37] that showed that the root bark of several African plants from the family Rubiaceae possess anti-inflammatory and analgesic effects. When compared to the standard ketoprofen, a significantly (p < 0.05) higher inhibition of carrageenan induced inflammation was exhibited by the extracts in the fourth and fifth hours (77.83 to 93.35%) for all the extracts except for n-hexane which had a reduced inhibition of oedema in the fifth hour. This shows that while ketoprofen may act by inhibiting the secretion of serotonin, histamine and bradykinin, the extracts produce its anti-inflammatory effect in the second phase by limiting production of prostaglandin and cyclooxygenase (COX-2), making Feretia apodanthera a promising plant for further studies. This is in line with the studies that reveals that formation of oedema caused by carrageenan is in two phases; the first hour after carrageenan injection (first or early phase) involves the release of serotonin, histamine and bradykinin while the second or late phase (2–5 h) with increased oedema formation that remains up to the fifth hour involves the release of prostaglandins [38] COX-2 also induction in the hind paw [39].

For FTIR, the analysis time was less than five minutes and it required a minute quantity of the sample. The spectrum revealed the presence of functional groups such as aromatics, ethers, esters, alcohol, alkanes, ketones, alkenes, alkynes, amines and carboxylic acid [40]. Fink [41] in his report stated that the presence of keto carboxylate elicits antioxidant effect by scavenging hydrogen peroxide;

| Table 5 Functional Groups Identified by FTIR analysis of F. apodanthera Ethanol Extract |
|---------------------------------------------------------------|
| S/No | Absorption peak (cm⁻¹) | Functional group | Assignment |
| 1    | 446.54                 | R-I stretch     | Alkyl halides |
| 2    | 563.23                 | C-S stretch     | Disulfides    |
| 3    | 1075.35                | C-O-C           | Cyclic ethers, large rings, C-O stretch |
| 4    | 1282.71                | P=O stretch     | Organic nitrates or phosphates |
| 5    | 1434.12                | -CH₃            | Methyl bend   |
| 6    | 1642.44                | C=C stretch     | Alkenes       |
| 7    | 2112.12                | C≡C stretch     | Alkyne        |
| 8    | 2348.41                | C≡O             | Ketones       |
| 9    | 2933.83                | >CH₂, C-H       | Alkane        |
| 10   | 3435.34                | R-C(O)-OH       | Carboxylic acid |
| 11   | 3938.77                | N-H stretch     | Amine         |

| Table 6 GCMS Profiling of the Ethanol Extract of Feretia apodanthera |
|---------------------------------------------------------------|
| PK | RT      | % Area | Trend | Compound name                              | CAS        |
|----|---------|--------|-------|--------------------------------------------|------------|
| 1  | 5.4176  | 1.2033 | 1      | 2-Pentanone, 4-Hydroxy-4-methyl-           | 000123–42-2 |
| 2  | 50.2953 | 0.1749 | 3      | Hexadecanoic acid, methyl ester            | 000112–39-0 |
| 3  | 56.157  | 0.686  | 3      | trans-9-Octadecenoic acid, pentyl ester    | 1,000,405–19-1 |
| 4  | 56.9995 | 0.109  | 2      | Methyl cyclohexanepropionate               | 020681–51-0 |
| 5  | 60.9195 | 4.323  | 4      | Palmitoleic acid                           | 000373–49-9 |
| 6  | 62.3116 | 0.6729 | 3      | 10-Undecenoyl chloride                      | 038460–95-6 |
| 7  | 90.3374 | 92.8309| 7      | Heptasiloxane, 1,1,3,3,5,5,7,7,9,9,11,11,13,13-tetradecamethyl- | 019095–23-9 |
compounds with aromatic rings and alcohol groups as seen in phenols have been known to modulate inflammation at different levels by decreasing the production of reactive nitrogen and oxygen species, limiting the activity of iNOS and COX, suppressing inflammatory chemokines and cytokines synthesis as well as controlling pathways for NF-kB signaling [42].

GCMS analysis of Feretia apodanthera showed the presence of 2-Pentanone, 4-hydroxy-4-methyl-; Hexadecanoic acid, methyl ester; trans-9-Octadecenoic acid, pentyl ester; Methyl cyclohexanepropionate; Palmitoleic acid; 10-Undecenoyl chloride and Heptasiloxane, 1,1,3,3,5,5,7,7,9,9,11,11,13,13-tetradecamethyl- (Table 6). Hexadecanoic acid, methyl ester for example, is reported to reduce the production of nitric oxide- a key mediator in inflammation processes [35] - in cells by inhibiting arachidonic acid production of nitric oxide- a key mediator in inflammation processes [35]- in cells by inhibiting arachidonic acid (Table 7) which is a precursor in the biosynthesis of prostaglandins and therefore, inhibit the levels of pro-inflammatory mediators such as interleukin-10 (IL-10), prostaglandin (PGE2) and tumor necrosis factor-alpha (TNFα) [43]. Palmitoleic acid which was also identified in the extract was shown to have inflammatory lowering effects [44]. Heptasiloxane, 1,1,3,3,5,5,7,7,9,9,11,11,13,13-tetradecamethyl- have been identified in highly medicinal plants such as moringa leaves [45].

### Conclusion

These results suggest that the extract of Feretia apodanthera may possess some antioxidiant properties and provide relief against inflammation making it a possible future therapy for inflammation. This may be the reason for the use of the extract in folklore medicine.

### Abbreviations

COX-2: Cyclooxygenase 2; DPPH: 1,1-diphenyl-2-picrylhydrazyl; F. apodanthera: Feretia apodanthera; FTIR: Fourier transform infrared; GCMS: Gas chromatography mass spectrometry; IAR: Institute for Agricultural Research; IC50: 50% Inhibitory concentration; IL: Interleukin; iNOS: Inducible nitric acid synthase; NF-κB: Nuclear factor kappa beta; STAT 1: Signal transducer and activator of transcription 1; TNF α: Tumour necrosis factor alpha; VEGF: Vascular endothelial growth factor

| S/No | Name of Compound | Molecular formula | Activity |
|------|------------------|-------------------|----------|
| 1    | 2-Pentanone, 4-hydroxy-4-methyl- | C₆H₁₂O₂ | 17-beta-hydroxysteroid dehydrogenase inhibitor, aryl-hydrocarbon-hydroxylase inducer, testosterone-hydroxylase-inducer, catechol-o-methyl-transferase inhibitor, methyl donor, methyl guanidine inhibitor |
| 2    | Hexadecanoic acid, methyl ester | C₁₇H₃₄O₂ | Acidulant, Acidifier, Arachidonic acid inhibitor, increase aromatic amino acid decarboxylase activity, inhibit production of uric acid, urinary-acidulant |
| 3    | trans-9-Octadecenoic acid, pentyl ester | C₂₃H₄₆O₂ | Arachidonic acid inhibitor, acidifier, acidulant, increase aromatic amino acid decarboxylase activity, catechol-o-methyl-transferase inhibitor, decrease glutamate oxaacacetate transaminase |
| 4    | Methyl cyclohexanepropionate | C₁₀H₁₈O₂ | catechol-o-methyl-transferase inhibitor, methyl donor, methyl-guanidine inhibitor |
| 5    | Palmitoleic acid | C₁₆H₃₀O₂ | Acidulant, Acidifier, Arachidonic acid inhibitor, increase aromatic amino acid decarboxylase activity, inhibit production of uric acid, urinary-acidulant |
| 6    | 10-Undecenoyl chloride | C₁₁H₁₉ClO | – |
| 7    | Heptasiloxane, 1,1,3,3,5,5,7,7,9,9,11,11,13,13-tetradecamethyl- | C₁₄H₄₄O₂Si₇ | – |

IC50: 50% Inhibitory concentration; IL: Interleukin; iNOS: Inducible nitric acid synthase; NF-κB: Nuclear factor kappa beta; STAT 1: Signal transducer and activator of transcription 1; TNF α: Tumour necrosis factor alpha; VEGF: Vascular endothelial growth factor

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### Availability of data and materials

The datasets used and analysed during this study are available from the corresponding author on reasonable request.

### Authors' contributions

OOO managed the literature searches, the lab work and the analysis of the study; DBJ designed the work, supervised it and edited the final manuscript; IS provided minor supervision; OOF managed the FTIR and GCMS aspect of this research; BTA and BK provided assistance in the antioxidant determination section of the experiment. All authors read and approved the final manuscript and are accountable for the integrity of this manuscript.

### Ethics approval and consent to participate

This research was done in compliance with the revised animals (Scientific and Biochemical Ethics Committee) of the Faculty of Science, Ahmadu Bello University, Zaria, Kaduna State, Nigeria.

### Consent for publication

Not applicable. The article does not include a clinical study.

### Competing interests

The authors declare that they have no competing interests. This manuscript is original and has not been submitted for possible publication to another journal neither has it been published elsewhere.

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