Biodecolorization of methyl orange by mixed cultures of brown-rot fungus Daedalea dickinsii and bacterium Pseudomonas aeruginosa

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Manuscript received: 27 February 2020. Revision accepted: 29 April 2020.

Abstract. Purnomo AS, Mawaddah MO. 2020. Biodecolorization of methyl orange by mixed cultures of brown-rot fungus Daedalea dickinsii and bacterium Pseudomonas aeruginosa. Biodiversitas 21: 2297-2302. This study investigated on the decolorization of methyl orange (MO) by mixed cultures of brown-rot fungus (BRF) Daedalea dickinsii and bacterium Pseudomonas aeruginosa. P. aeruginosa was added into D. dickinsii culture at 2, 4, 6, 8, 10 mL (1 mL = 5.05 × 10^12 CFU). All of mixed cultures had ability to decolorize MO (final concentration 100 mg/L) in potato dextrose broth (PDB) medium for 7 days incubation. The addition of 4 mL of P. aeruginosa showed the highest MO biodecolorization approximately 97.99%, while by D. dickinsii only was 67.54%. Cr(H2N-O-S); Cr(H2N-O-S); and Cr(H2N-O-S) were identified as MO metabolites. This study indicated that mixed cultures of D. dickinsii and P. aeruginosa have great potential for high efficiency, fast and cheap dye wastewater treatment.

Keyword: Biodecolorization, Daedalea dickinsii, methyl orange, Pseudomonas aeruginosa

INTRODUCTION

The activities of the textile industry cannot be separated from the dyed process and produce waste residue dyes that are discharged into the environment. Wastewater containing textile dyes can be toxic, carcinogenic and may even cause genetic and harmful mutations for aquatic organisms and humans. One of azo's textile dyes is methyl orange (MO), which is a water-soluble dye, used as a textile dye, and as an indicator pH (Ljubas et al. 2015). The release of industrial pollutants has become a major concern for human health and the environment, which dyes are amongst the major contributors to environmental pollution. Dyes are used extensively to alter the color characteristics as well as enhance the appearance of various products. However, as most natural dyes are unstable, thus synthetic dyes have emerged as an essential alternative (El Nemr 2012). In the staining process, 15% of the MO dyes are not absorbed, released, and flowed in wastewater streams. Even at very low concentrations, MO dye in water is very visible. From an ecological perspective, the presence of azo dyes in textile effluents is highly undesirable as azo dye effluents are heavily colored, concentrated with salt, and have high biological oxygen demand (BOD) and chemical oxygen demand (COD). BOD and COD values in some textile effluents were found to be in the range of 220-490 mg/L and 180-940 mg/L respectively (Chiong et al. 2016). The contamination may inhibit the penetration of light into the water and may invade the process of photosynthesis by aquatic organisms. Therefore, decolorization is a major problem in wastewater treatment from industry (Liu et al. 2012).

Conventional treatment of dye waste has been widely applied such as chemically, physically, and biologically. Over the past decade, researchers have focused more on biological treatment because it is cheap and environmentally friendly. Among the many biological agents involved in bioremediation, bacteria, and fungi are important agents (Ali and Ahmed 2010). The dye degradation of MO using the fungus has been much studied. The commonly used fungus is a type of white-rot fungus, although many other types of fungus are reported to degrade azo-colored dyes such as MO (Bumpus 2004). Only a few studies reported degradation or decolorization using brown-rot fungi (BRF) as reported by Ali and Hameed (2010). Aspergillus flavus SA2 can degrade dye Red Acid 151 by 67% (initial concentration of 20 mg/L). While the fungus Penicillium spp. and Alternaria spp. SA4 can degrade orange II dye by 34% and 57% (initial concentration 20 mg/L; Bumpus 2004). Since BRF does not have ligninolytic enzymes, it has been proposed that they use hydroxyl radicals produced via the Fenton reaction for the degradation of wood components (Purnomo et al. 2010c, 2011a). In BRF, extracellular Fenton-type mechanisms have been reported to be involved in degradation of several xenobiotic compounds, including the fluoroquinolone antibiotics enrofloxacin and ciprofloxacin, polyethylene glycol, chlorophenol, 2,4,6-trinitrotoluene (Purnomo et al. 2011b), aldrin, dieldrin (Purnomo et al. 2017a, Purnomo 2017), heptachlor, heptachlor epoxide (Purnomo et al. 2013, 2014) and methylene blue dye (Rizqi and Purnomo 2017). Degradation of MO by brown-rot fungus Daedalea dickinsii fungi was evaluated in PDB (Potato Dextrose Broth) at a concentration of MO 75 mg/L of 97.56% within 14 days incubation (Purnomo et al. 2019b). Although D. dickinsii had high ability to degrade MO, degradation time consumed a long time, thus culture modification is needed to improve the ability of D. dickinsii.
Several studies have been reported on the use of bacterial agents to degrade azo dyes, in which two enzymes play a role in the biodecolorization of azo dyes: azoreductase and laccase (Singh et al. 2015). Pseudomonas species bacteria have been reported capable to decolorize dyes. The novel isolated laccase producing *Pseudomonas stutzeri* MN1 has ability to decolorize congo red and gentian violet (Kuppusamy et al. 2017). Besides, *Pseudomonas aeruginosa* has been reported to be able to decolorize remazol orange dye with an initial concentration of 200 mg/L within 24 hours at 82.4% (Sarayu and Sandhya 2010). *P. aeruginosa* is a Gram-negative, rod-shaped, asporogenous, and monoflagellated bacterium. *P. aeruginosa* grows well at 25°C to 37°C, and its ability to grow at 42°C helps distinguish it from many other *Pseudomonas* species. This suggests that *P. aeruginosa* can be used as biodecolorization agent. Some studies suggest that mixed cultures can improve the ability of culture degradation. The addition of *P. aeruginosa* has been reported enhance DDT degradation by *Pleurotus ostreatus* (Purnomo et al. 2017), and *Fomitopsis pinicola* (Sariwati et al. 2018). Besides, mixed cultures between *D. dickinsii* and *P. aeruginosa* have been used as DDT degradation agents, in which the addition of 10 mL (1 mL = 1.05 x 10^8 CFU/mL) of *P. aeruginosa* can degrade 100% DDT for 7 days incubation (Setyo et al. 2018). It proved that mixed cultures of *D. dickinsii* and *P. aeruginosa* is a potential degradation agent that can be used to decolorize MO dyes. Given these properties, the ability of mixed cultures of *D. dickinsii* and *P. aeruginosa* to decolorize MO was investigated and the metabolic products were identified.

**MATERIALS AND METHODS**

Materials

Brown-rot fungus *Daedalea dickinsii* NBRC 31163 and bacterium *Pseudomonas aeruginosa* NBRC 3009 were a collection from Microbial Chemistry laboratory of Department of Chemistry, Institut Teknologi Sepuluh Nopembor (ITS), Surabaya, Indonesia. Methyl orange (MO) textile dye was purchased from SAP Chemicals (96% purity). Growth media were Nutrient Agar (NA, Merck, Germany), Nutrient Broth (NB, Merck, Germany), Potato Dextrose Broth (PDB, Himedia, India), and Potato Dextrose Agar (PDA; Merck, Germany). Aqua DM and ethanol 70% were purchased from PT. Sumber Ilmiah Persada Indonesia.

Fungus and bacterium culture

Stock cultures of *D. dickinsii* NBRC 31163 were maintained on PDA plates that had been incubated at 30°C in 7 days. The mycelia from the agar plate were transferred to a sterile blender cup containing 25 mL of sterile water and then homogenized for 30 sec. One milliliter of this homogenate was inoculated into 8 mL of PDB medium in a 100-mL Erlenmeyer flask. The cultures were pre-incubated statically at 30°C for 7 days (Purnomo et al. 2010a). Besides, Bacterium stock cultures of *P. aeruginosa* NBRC 3080 were maintained on NA that had been incubated at 37°C. The colony was inoculated into 10 mL of NB medium in 50-mL Falcon flasks. The cultures were pre-incubated at 37°C for 24 hours (Wahyuni et al. 2017).

Biodecolorization MO by fungus *Daedalea dickinsii*

After pre-cultivating for 7 days, 10 mL of PDB medium was added into inoculated fungus cultures (final volume 20 mL), and MO (final concentration 100 mg/L) was added to each fungus-inoculated flask. The cultures were further incubated for 7 d at 30°C (Purnomo et al. 2010b; Setyo et al. 2018).

Biodecolorization MO by bacterium *Pseudomonas aeruginosa*

After pre-cultivation for 24 h, *P. aeruginosa* cultures were inoculated into the PDB medium at 2, 4, 6, 8 and 10 mL (1 mL = 5.05 x 10^12 CFU, ultimate volume 20 mL). Every bacterium inoculated flask was added with MO (final concentration 100 mg/L). The cultures were cultivated for 7 d at 30°C (Sariwati et al. 2017).

Biodecolorization MO by mixed cultures of *Daedalea dickinsii* and *Pseudomonas aeruginosa*

Biodecolorization of MO by mixed cultures was performed by adding 2, 4, 6, 8 and 10 mL (1 mL = 5.05 x 10^12 CFU) of pre-incubated bacteria into Erlenmeyer flask containing 9 mL of pre-incubated fungus following by addition of PDB to the total volume of 20 mL. Each culture was added to MO (final concentration of 100 mg/L). The culture was incubated statically for 7 days at 30°C. The synergistic relationship of mixed cultures was expressed with Ratio Optimization (RO) that calculated as the amount of decolorization by mixed cultures per total amounts of decolorization by fungus and bacteria (Purnomo et al. 2019a).

Analytical method and identification of metabolites

After the incubation process, cultures were separated by centrifuge (4000 rpm for 15 min). The resulting supernatant was measured its absorbance by a UV-Vis spectrophotometer (Purnomo et al. 2017). For abiotic control, PDB was added MO reach to 100 mg/L concentration without the addition of cultures. The percentage of MO decolorized was calculated by:

\[
\% \text{ decolorization} = \frac{A_c - A_t}{A_c} \times 100\%
\]

Where: Ac is control absorbance, and At is treatment absorbance (Rizqi and Purnomo 2017). The identification of metabolites product from MO degradation was performed by analyzing the supernatants using a liquid chromatography–time of flight mass spectrometry (LC-TOF/MS). The ionization source was ionizing electrospray (ESI) with a mass range of 50-1000. The gradient elution method was used with flow rate of 0.2 mL/min in the first three minutes and the next seven minutes using flow rate of 0.4 mL/min. The phase of motion was used methanol and water with a ratio of 99:1 in the initial three minutes and 61:39 for the remaining seven minutes. The column was Acclaim TM RSC 120 C18 type column with size 2.1x100 mm and column temperature 33°C (Boelans and Purnomo 2019).
RESULTS AND DISCUSSION

Biodecolorization MO by fungus

The absorbance profile of MO during degradation by BRF *D. dickinsii* was shown in Figure 1. MO maximum wavelength was detected at 465 nm, in which the absorbance of abiotic control and treatment by fungus was 3.424 and 1.109 respectively. These results showed that the decrease in the value of absorbance at 465 nm indicated of decolorization of MO by *D. dickinsii*. However, the peak of MO was shifted from 465 nm to 510 nm after incubation due to acidic conditions in culture that lead protonation process of MO and cause shifting the peak of the chromophore, thus change color from orange to red (Purnomo et al. 2019b). During pre-incubation, fungus produces some organic acids in which the pH culture was 1.9.

Based on the absorbance result, % decolorization was calculated, in which *D. dickinsii* decolorized MO approximately 68%. The ability of *D. dickinsii* to degrade MO might be associated with the ability of this fungus to produce hydroxyl radicals generated by the Fenton reaction during the incubation (Purnomo et al. 2010). Besides, MO may also be degraded by extracellular enzymes, as some fungi produce some degradative enzymes, such as laccases and peroxidases (Singh et al. 2015). The more extracellular hydroxyl radicals and enzymes are produced during incubations (Kaneko et al. 2005), The highest decolorization was obtained.

Biodecolorization MO by bacterium

MO biodecolorization profile by bacteria was shown in Figure 2. The highest peak of MO decolorization by the bacterium was also obtained at 465 nm. The MO decolorization by higher volume of bacteria resulted in decreasing of MO peak which indicating that MO was further decolorized. Azoic dye decolorization was occurred under anaerobic conditions by bacteria, which requires organic carbon as an energy source. Decolorization might be attributed to non-specific extracellular reactions occurring between reduced compounds generated by the anaerobic bacteria (Pandey et al. 2007). Table 1 provides the percentage of decolorization (% decolorization) of the bacterium during degradation. The bacterium decolorized MO by approximately 50%, 90%, 91%, 91% and 92% at 2, 4, 6, 8, dan 10 mL of bacteria in PDB medium, respectively. It showed that decolorization of MO was not significantly different in variation volume bacterium of 4, 6, 8, dan 10 mL. The addition of all variations of bacteria showed increasing decolorization of MO.

Biodecolorization percentages of mixed cultures were shown in Table 1. As the volume of bacteria increases, the percentage of decolorization increases, except in the addition to 10 mL. Optimal decolorization has occurred in addition to bacteria of 4 mL. On the other hand, the optimization ratio (OR) indicated the level of enhancement of MO degradation due to the synergistic relationship between *D. dickinsii* and *P. aeruginosa*, compared with the degradation by the individual organisms. The addition of 2 mL of *P. aeruginosa* showed the highest RO, which enhanced the degradation by approximately 2 times. The addition of 10 mL of *P. aeruginosa* showed the lowest RO. The optimal decolorization was obtained at mixed cultures of *D. dickinsii* with the additional of 4 mL of *P. aeruginosa*, thus the addition of bacteria *P. aeruginosa* into fungus *D. dickinsii* culture can increase the ability of decolorization by 34.24% (from 64.41% by *D. dickinsii* culture only to 98.65% by mixed cultures). Mixed microbial cultures have more power to degrade pollutants because they have more genetic information to produce complex enzymes and metabolites (Grizca and Setyo 2018).

The identification of metabolites

Based on characterization by LC-TOF/MS, four metabolites were detected at time retention 1.29; 2.27; 2.7; 3.95; and 7.29 mins. Identification of metabolites was determined based on similarity between MS spectrum and time retention from database (Figure 4). The result of LC analysis showed that the mixed cultures were able to transform MO to some metabolites. Compared with control, MO was identified at time retention 7.29 mins, the MO peak was decreased compared with MO peak control. Based on TOF/MS data, the peak at retention time 1.29 min has m/z 374 which identified as 4-[N-(4-diethylamino-3-hydroxy-cyclohexa-2,5-dienylidene)hydrazino]-2-methoxy-benzenesulfonic acid (*C₁₅H₁₀N₂O₅*S). The compound was reported previously which had fragments in m/z 149 in the form of compound *C₆H₇N₂* (N, N-diethylamine fenildiazin), which is in accordance with previous research on the degradation of azo dyes by laccase enzyme from WRF, and there are m/z 118 fragments, *C₅H₄N₂* was 4-(methylamino) aniline (Mishra et al. 2011).

At a peak of 2.27 mins, metabolite was identified as 4-[N-(4-diethylamino-3-methoxy-cyclohexa-2,5-dienylidene)hydrazino]-2-methoxy-benzenesulfonic acid (*C₁₆H₁₂N₂O₅*S) with m/z 338. This compound is supported by the presence of fragments in MS data with m/z 213 as *C₁₃H₁₀N₃* fragment m/z 137 *C₅H₄N₂* (1,4-diamine) in accordance with research on metabolites produced by *Aeromonas* sp. in MO (Du et al. 2015). The peak of 2.71 mins was identified as 4-[N-(4-diethylamino-3,5-dimethoxy-cyclohexa-2,5 -dienylidene)hydrazino]-2-methoxy-benzenesulfonic acid (*C₁₇H₂₂N₄O₅*S) with m/z 434. This compound is supported by fragments m/z 213, m/z 213, and m/z 152.
Table 1. Percentage of MO decolorization by bacteria culture (Pseudomonas aeruginosa) and mixed cultures (Daedalea dickinsii and P. aeruginosa)

| Amount of bacteria culture (mL) | P. aeruginosa alone | Mixed cultures | Optimization Ratio |
|---------------------------------|---------------------|----------------|-------------------|
| 0                               | 0 ± 0.016           | 0.00 ± 0.017   |                   |
| 2                               | 49.61 ± 0.090a      | 94.29 ± 0.004a | 0.8a              |
| 4                               | 90.28 ± 0.040b      | 97.99 ± 0.011b | 0.62b             |
| 6                               | 91.16 ± 0.035b      | 96.41 ± 0.003b | 0.61b             |
| 8                               | 91.41 ± 0.029b      | 92.26 ± 0.015b | 0.58b             |
| 10                              | 91.46 ± 0.016b      | 87.35 ± 0.010b | 0.55c             |

Analyses were conducted by spectrophotometer UV-VIS. Data are mean ± standard deviation (n=3). A 1 mL of bacteria ≈ 5.05 × 10^{12} CFU. Data followed by the different minor letter on each column indicates significantly different (P < 0.05).

Figure 1. The profile of MO biodecolorization by Daedalea dickinsii

Figure 2. The profile of MO biodecolorization by bacteria cultures

Figure 3. The profile of MO biodecolorization by mixed cultures

Figure 4. LC chromatogram of metabolite products of MO degradation by mixed cultures (red chromatogram is MO control; green chromatogram is treatment)
The last metabolite was identified at 3.95 mins as 4-[N-(4-dimethylamino-3,5-dimethoxy-cyclohexa-2,5-dienylidene)-hydrazino]-2-methoxy-benzene sulfonylic acid (C₁₅H₁₇N₅O₆S) with 391 m/z. This compound is supported by fragments m/z 278 and fragments m/z 171 allowed the presence of compound C₁₅H₁₇SNO₃ (4-sulfonic acid) (Hao et al. 2016). Based on the identification of metabolites, the MO decolorization pathway by mixed cultures of D. dickinsii and P. aeruginosa cultures were proposed in Figure 5. The pathway of degradation starts from MO structure with m/z 327 and then MO had initial degradation in its benzene ring which begins to undergo the process of the methylation and oxidation to form 4-[N-(4-dimethylamino-3-hydroxy-cyclohexa-2,5-dienylidene)-hydrazino]-2 methoxy benzene sulfonylic acid with m/z 374 with additional functional group. After that changed, the alcohol functional group become methoxy and forming 4-[N-(4-dimethylamino-3-methoxy-cyclohexa-2,5-dienylidene)-hydrazino]-2-methoxy- benzene sulfonylic acid (C₁₆H₁₈N₅O₈S) with m/z 388. The MO structure becomes bulkier with the change on its benzene ring with increasing polarity and affinities of its structure. The last form 4-[N-(4-dimethylamino-3,5-dimethoxy-cyclohexa-2,5-dienylidene)-hydrazino]-2-methoxy-benzene sulfonylic acid (C₁₅H₁₇N₅O₆S) with 391 m/z as results of the degradation pathway.

**ACKNOWLEDGEMENTS**

This study was supported by a grant from research project from the Directorate of Research and Community Service, Directorate General of Strengthening Research and Development, Ministry of Research, Technology and Higher Education No. 5/E1/KP.PTNBH/2019.

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