Structural basis for the inhibition of voltage-dependent K⁺ channel by gating modifier toxin

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Voltage-dependent K⁺ (Kᵥ) channels play crucial roles in nerve and muscle action potentials. Voltage-sensing domains (VSDs) of Kᵥ channels sense changes in the transmembrane potential, regulating the K⁺-permeability across the membrane. Gating modifier toxins, which have been used for the functional analyses of Kᵥ channels, inhibit Kᵥ channels by binding to VSD. However, the structural basis for the inhibition remains elusive. Here, fluorescence and NMR analyses of the interaction between VSD derived from KᵥAP channel and its gating modifier toxin, VSTx₁, indicate that VSTx₁ recognizes VSD under depolarized condition. We identified the VSD-binding residues of VSTx₁ and their proximal residues of VSD by the cross-saturation (CS) and amino acid selective CS experiments, which enabled to build a docking model of the complex. These results provide structural basis for the specific binding and inhibition of Kᵥ channels by gating modifier toxins.

Voltage-dependent K⁺ (Kᵥ) channels alter their K⁺-permeability across membrane lipid bilayers in a membrane potential-dependent manner, playing crucial roles in nerve and muscle action potentials¹. The Kᵥ channels function as a tetramer, in which each subunit possesses six transmembrane helices, S₁–S₆. Tetrameric assembly of the S₅–S₆ regions of the four subunits (referred to as a pore domain) forms a pore for the K⁺-permeation, in which a crossing of the four S₆ helices at the intracellular side of the pore (referred to as a helix bundle crossing) acts as a gate to physically preclude the K⁺-permeation. The opening and closing of the gate (gating) is allosterically regulated by voltage-sensing domains (VSDs) comprised of the S₁–S₄ helices that are located at the periphery of the pore domain²,³. A number of positively charged residues of S₄ are responsible for the membrane potential-dependent S₄ shift⁴–¹⁰; at the resting potential, S₄ shifts to the intracellular side of the membrane (‘down’ conformation), whereas during depolarization, S₄ shifts to the extracellular side (‘up’ conformation). This voltage-dependent conformational change of VSD is assumed to cause the gating.

To date, a variety of peptide toxins that inhibits specific Kᵥ channels have been isolated from venomous organisms such as snakes, scorpions and spiders, and used for the characterization of the Kᵥ functions¹¹. These toxins can be classified into two groups, a pore blocking toxin and a gating modifier toxin. Pore blocking toxins target the extracellular side of the pore domain, and the structural basis, on which the toxins physically occlude the pore, has been revealed¹²–¹⁵. On the other hand, gating modifier toxins bind to VSD, and are assumed to alter the conformation and energetics of voltage-dependence of VSD¹⁶–¹⁸ whereas the structural basis for the inhibition has not been fully elucidated.

Recently, the structures of several gating modifier toxins targeting Kᵥ channels such as VSTx₁, SGTx₁ and HaTx, have been determined¹⁹–²¹. These toxins commonly possess a cluster of solvent-exposed hydrophobic residues (referred to as a hydrophobic patch) surrounded by highly polar residues, enhancing the affinity for their target Kᵥ channels by allowing the toxins to partition into the membrane²²–²⁵. However, mutagenic study reported that the hydrophobic patch of SGTx₁ also plays a critical role in the recognition of its target, Kᵥ2.1, in the membrane²⁶.

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In addition, VSTx1 reportedly inhibits an archaebacterial \( K \) channel, \( K_{\alpha}AP \), where VSTx1 exclusively binds to the VSD and the pore domain is not required for the toxin-channel interaction\(^2^\). Furthermore, electro physiological studies suggested that the \( K_{\alpha}AP \) is inhibited upon depolarization by recognizing the up conformation of VSD\(^3^\). However, no structure of VSD in complex with a gating modifier toxin has been reported, and thus it remains unknown how these toxins prevent the voltage-dependent conformational change of VSD.

In this study, we performed the fluorescence and NMR analyses of the interaction of VSTx1 and VSD derived from \( K_{\alpha}AP \), indicating that VSTx1 stabilizes the up conformation of VSD. In addition, we identified the VSD binding residues of VSTx1 and their proximal residues of VSD by the cross-saturation (CS)\(^2^9,3^0\) and amino acid selective CS (ASCS)\(^3^1\) experiments. Based on these results, we built a docking model of VSTx1 and VSD, providing the structural basis for the specific binding and the inhibitory mechanism of \( K \) channels by gating modifier toxins.

**Results**

**Characterization of the prepared VSD and VSTx1 proteins.** VSD from \( K_{\alpha}AP \) (residues 12 to 136, the residue numbers correspond to those in the crystal structure\(^3^2\)) solubilized in n-decyl-\( \beta \)-D-maltopyranoside (DM) micelles was purified to homogeneity (Supplementary Fig. S1). Size exclusion chromatography analysis indicated that the prepared VSD exists as a monomer in the DM micelles (Supplementary Fig. S1). While \( ^1H,^{15}N \) TROSY NMR spectrum of uniformly \( ^2H,^{15}N \)-labeled VSD exhibits a good dispersion of the signals in a \( ^1H \)-dimension, some signals are broadened, suggesting that VSD is globally folded with some residues undergoing chemical exchange (Supplementary Fig. S1).

On the other hand, VSTx1 was expressed in \( E. coli \) and purified to homogeneity (Supplementary Fig. S2). We assigned backbone NMR resonances of the uniformly \( ^13C,^{15}N \)-labeled VSTx1 by a series of triple resonance experiments (Supplementary Table S1 and Supplementary Fig. S2). Chemical shift index analysis confirmed that the secondary structure of the prepared VSTx1 is consistent with the previously reported solution structure\(^2^1\) (Supplementary Fig. S2).

**Voltage-dependent conformational change of VSD and its inhibition by VSTx1.** In order to investigate the voltage effect on the VSD conformations, a fluorescence-labeled VSD, in which monobromobimane (mBBr) was chemically attached to a Cys residue mutated from Val 119 that lies at the extracellular end of S4, was reconstituted into liposomes. This assay was originally developed to study \( H^+ \) flux through V-ATPase\(^3^3\), and then applied to study voltage-dependent conformational change of \( H^+ \) channel and VSD\(^3^4\). Specifically, liposomes were prepared in the presence of 150 mM \( K^+ \) and diluted into buffer containing lower concentration of \( K^+ \), which generated the \( K^+ \) gradient across the membrane. By adding a \( K^+ \)-selective ionophore, valinomycin, the \( K^+ \) efflux produces the membrane potential (negative inside relative to outside). When the 'up' to 'down' conformational change of the S4 helix of the reconstructed VSD with its N- and C-termini inside the liposome is caused by the formation of the membrane potential, the intensity of the fluorescence derived from mBBr, which changes the environment from the one exposed to the solvent to the one buried in the membrane, would increase. On the other hand, the VSD with its N- and C-termini outside the liposome (corresponding to the opposite orientation) would experience positive voltage upon the formation of the membrane potential, where the S4 helix would remain in the up conformation, and mBBr would remain exposed to the solution. Therefore, molecules in this orientation are not supposed to contribute to the "change" in the fluorescence intensity.

Figure 1a shows transient increases and subsequent decays of the fluorescence intensities upon formation of the various membrane potentials, suggesting that the mBBr moiety on the extracellular end of S4 buried in the membrane transiently and returned to the initial exposed position. The baseline potential before the addition of valinomycin was confirmed to be 0 mV since the fluorescence intensity of the diluted liposomal solutions was equivalent to that of the liposomal solution suspended in the same buffer as the one used for the liposome preparation. The fluorescence decays might reflect the 'down' to 'up' conformational change of the VSD caused by the depolarization of the membrane due to the \( H^+ \) entry through an intrinsic proton conduction pathway of VSD, as described in the previous report\(^3^4\).

Voltage-dependence of the transient fluorescent increase as shown in Fig. 1c (rhombic) suggests that VSD adopts mostly up and down conformations at 0 and \(-150\) mV, respectively, with the half maximal value at ca. \(-90\) mV. This half maximal value of the VSD movement was 30 to 40 mV smaller than a typical mid-point potential of the conductance-voltage curve of \( K_{\alpha}AP \) in POPE/POPG membrane\(^2^8,3^2\), which is consistent with the differences in the mid-point potentials between the VSD movements and the conductance of \( K_{\alpha}AP \) and other \( K \) channels\(^4^5,3^5\). Therefore, we conclude that the isolated VSD prepared here exhibits the voltage-dependent conformational change as previously reported.

Then, we investigated the effect of VSTx1 on the voltage-dependent fluorescent change of the mBBr-labeled VSD. Figure 1b shows the fluorescence upon the formation of the various membrane potentials in the presence of VSTx1 and Fig. 1c (square) shows the voltage-dependence of the transient fluorescence increase, clearly indicating that VSTx1 suppressed the fluorescence change of the S4-attached mBBr. Furthermore, no significant change was observed for the fluorescence profile upon addition of VSTx1 without the polarization (Fig. 1d). These results strongly suggest that VSTx1 binds to VSD in the absence of membrane potential, stabilizes the up conformation of VSD without significant structural change around S4, and inhibits the conformational change to the down conformation.
VSD adopts an up conformation in DM micelles. In order to reveal the structural basis of the inhibition of the voltage-dependent conformational change of VSD by VSTx1, we investigated the interaction between VSD and VSTx1 in the DM micelles. First, we examined whether the VSD in the DM micelles adopts an up or down conformation, based on solvent accessibilities of Cys residues introduced to the 11 positions of VSD from S3 to S4 (Fig. 2). These results indicate that water-soluble Cys modification...
reagent, Mal-PEG, reacted with the mutants, H109C, A111C, L118C and V119C, in which the former two mutation sites are on the C-terminal region of S3 (referred to as S3b) and the latter two are on the N-terminal region of S4. Based on the report that residues on S3b and the N-terminus of S4 are exposed to the extracellular solution in the up conformation whereas C-terminal residues of S4 are exposed to the intracellular solution in the down conformation, our observations of the solvent-exposed residues are consistent with the up conformation.

Direct binding of VSTx1 to VSD in DM micelles. We investigated the interactions of VSTx1 with DM and VSD by monitoring the NMR spectra of VSTx1. First, we observed a series of ¹H-¹⁵N HSQC spectra of uniformly ¹⁵N-labeled VSTx1 at the increasing concentrations of DM. A number of signals showed chemical shift changes, which were saturated at the DM concentration of 150 mM (Supplementary Fig. S3). Then, sequential additions of VSD into the VSTx1 solution in the presence of 150 mM DM also exhibited chemical shift changes (Fig. 3a). These results indicate the direct interactions of VSTx1 with DM and VSD. The backbone assignments of VSTx1 in complex with VSD in 150 mM DM were easily traced by sequential additions of DM up to 150 mM, followed by the additions of the VSD solution. The residue-specific analyses of the chemical shift changes of VSTx1 by DM and VSD were shown in Supplementary Figure S3.

The binding affinity between VSTx1 and VSD in the DM micelles was quantitatively analyzed by isothermal titration calorimetry (ITC). Fitting of the ITC isotherm resulted in the dissociation constant (K_d) of 1.5 ± 0.2 μM, ΔH: −4.5 ± 0.2 kcal mol⁻¹ and ΔS: 11.4 cal mol⁻¹ K⁻¹.

On the other hand, we compared ¹H-¹⁵N TROSY spectra of uniformly ²H, ¹⁵N-labeled VSD in DM micelles in the presence and absence of VSTx1 (Fig. 3b). An overlay of these spectra shows that VSTx1 caused chemical shift changes larger than the half width of the signals of VSD. However, only a limited number of signals of the VSD experienced the chemical shift change upon the VSTx1 binding, suggesting...
that the VSD conformation is not largely affected by the VSTx1 binding. Thus, the VSD seems to remain in an up conformation even in complex with VSTx1 in DM micelles, as observed in the lipid bilayer (Fig. 1d).

It should be noted that some NMR signals of VSD are broadened as stated above. The broadening was not improved even in the presence of VSTx1, which precluded complete assignments of the NMR resonances of VSD and NOE-based structure determination of the VSTx1-VSD complex. Here, we tried to obtain structural information of the VSTx1-VSD interaction through the NMR spectra of VSTx1 reflecting the interactions with VSD and DM without the assignments of the NMR resonances of the VSD.

DM and VSD binding residues of VSTx1 identified by the CS experiments. In order to reveal how VSTx1 recognizes VSD to inhibit KvAP, we applied cross-saturation (CS)\textsuperscript{31} and amino acid-selective cross-saturation (ASCS)\textsuperscript{31} methods. The former can identify VSTx1 residues binding to DM and/or VSD, and the latter can identify intermolecular proximal residue pairs between VSTx1 and VSD, both of which do not require the assignments of the NMR signals of VSD. The information on proximal residue pairs between VSTx1 and VSD enables docking of the reported structures of the two proteins since the individual structures remain essentially unchanged upon binding as suggested by the small chemical shift changes (Fig 3a,b and Supplementary Fig. S3). As the first step of this strategy for the structure determination of the VSTx1-VSD complex, we identified DM and VSD binding residues of VSTx1 by the CS method.

The CS method uses uniformly 2\textsuperscript{H}, 15\textsuperscript{N}-labeled VSTx1 in complex with unlabeled VSD in the DM micelles. Irradiation of the radio frequency pulses on 1\textsuperscript{H} resonances at 0 to 3.0 ppm saturates the DM and VSD resonances simultaneously. Through the dipolar-dipolar interactions, the saturation causes the signal intensity reduction of the 1\textsuperscript{H}-15\textsuperscript{N} TROSY signals of the VSTx1 residues located within 5 to 7 Å from the bound DM and/or VSD molecules (Fig. 4a,b).

Then, a control experiment was carried out by using uniformly 1\textsuperscript{H}-labeled VSD instead of the unlabeled one, in which the CS only from the DM molecules should be observed. Large intensity reductions were observed for Phe 5, Met 6, Trp 7, Lys 8, Cys 9, Asp 18, Trp 27, Cys 28 and Val 29 (Fig. 4c). These residues are clustered on the structure of VSTx1, indicating that this site is the DM binding site of VSTx1 in the VSTx1-VSD complex (Fig. 4e).

In order to evaluate the CS from VSD, the intensity reduction ratios obtained by using uniformly 2\textsuperscript{H}-labeled VSD are subtracted from those obtained by using unlabeled VSD (Fig. 4d). The differences in the reduction ratios, \(\Delta RR\), were significantly large for Val 20, Ser 22, Trp 25, Ser 32 and Phe 34, which are clustered on the opposite surface of the DM binding residues (Fig. 4f). Therefore, we concluded that the surface formed by these residues is the VSD binding site.

Intermolecular proximal residue pairs identified by the ASCS method. The ASCS method uses amino acid selectively 1\textsuperscript{H}-labeled VSD in a 2\textsuperscript{H}-background instead of unlabeled VSD in the CS method. Single ASCS experiment provides the information that cross-saturated VSTx1 residue(s) is proximal to any of the 1\textsuperscript{H}-labeled amino acid residues of VSD. When multiple ASCS results using a number of differentially labeled VSD are collected, combinatorial analyses can specify pairs of the CS-source residue(s) of VSD and the corresponding CS-acceptor residue(s) of VSTx1 based on their spatial complementarities on the protein surfaces\textsuperscript{31}.

We carried out three ASCS experiments, in which either Phe, Ile or Leu of VSD was selectively 1\textsuperscript{H}-labeled in a 2\textsuperscript{H}-background. The intensity reduction ratios of a control experiment using uniformly 2\textsuperscript{H}-labeled VSD (Fig. 4c) were subtracted from those from each ASCS results. As a result, Ser 22 of VSTx1 was cross-saturated from \([1\textsuperscript{H}-\text{Phe}]\) VSD, Val 20 from \([1\textsuperscript{H}-\text{Ile}]\) VSD, and Trp 25, Ser 32 and Phe 34 from \([1\textsuperscript{H}-\text{Leu}]\) VSD (Fig. 5a–c left).

These ASCS results, with reference to the individual structures of VSTx1 and VSD, can identify the VSTx1 binding residues of VSD. Since VSD possesses only two Phe residues, Phe 116 and Phe 124, either or both of these Phe residues should be close proximity to the VSTx1 residue that is cross-saturated from \([1\textsuperscript{H}-\text{Phe}]\) VSD, Ser 22. Similarly, either or some of 10 Ile residues of VSD should be close to Val 20 of VSTx1, and either or some of 28 Leu residues should be close to Trp 25, Ser 32 and Phe 34 of VSTx1, respectively (Fig. 5a–c right).

Systematically, all possible combinations of the CS-source residues were evaluated based on the maximum deviation of the CS-source residues and the cross-saturated amide hydrogen atoms (Supplementary Table S2). As a result, we identified five proximal residue pairs between VSTx1 and VSD, Val 20—Ile 127, Ser 22—Phe 124, Trp 25—Leu 121, Ser 32—Leu 128 and Phe 34—Leu 128 (Fig. 5d). All the identified VSD residues (Leu 121, Phe 124, Ile 127 and Leu 128) lie on the S4 helix, indicating that VSTx1 binds to S4 of VSD.

NMR-derived docking model of the VSTx1-VSD complex. Then, we used a docking software, HADDOCK\textsuperscript{37}, to build a docking model of the VSTx1-VSD complex that satisfies all the proximal residue pairs experimentally identified by the ASCS method. The obtained structure exhibited no significant violation or increase of energy values (Supplementary Table S3) and all the constraints obtained by the ASCS method were satisfied in the generated structure as below; Val 20 H\(^N\)—Ile 127 H\(\beta_1\) : 1.9 Å, Ser 22 H\(^N\)—Phe 124 H\(\beta_3\) : 4.7 Å, Trp 25 H\(^N\)—Leu 121 H\(\beta_1\) : 4.8 Å, Ser 32 H\(^N\)—Leu 128 H\(\beta_2\) : 1.7 Å, and Phe 34...
HN—Leu 128 Hδ1: 1.9 Å (Fig. 6). Therefore, we concluded that the structure of VSTx1-VSD complex was appropriately generated by using a docking with the NMR-derived structural constraints.

Discussion
Ruta & MacKinnon indicated that VSTx1 binding residues exclusively reside on the VSD and the pore domain is not involved. This study characterized the direct interaction between VSTx1 and the isolated VSD by fluorescence, ITC and NMR studies, indicating that VSTx1 binds to the VSD in an up conformation in lipid bilayer and DM micelles, which is consistent with the previous electrophysiological results. The dissociation constant ($K_d$) between VSTx1 and the VSD in the DM micelles was 1.5 μM, which is about three orders of magnitude larger than the previous observations in the membrane. The high affinity binding of VSTx1 and the VSD in membrane consists of two effects: partitioning of VSTx1 into the lipid bilayer (membrane-water partition coefficient of VSTx1 is reported to be at most $10^5$) and the direct binding of VSTx1 and the VSD. Our ITC study used the VSD and VSTx1 that are pre-reconstituted in detergent micelles. Therefore, our ITC experiments in detergent cannot reproduce membrane partitioning effect, which probably resulted in the lower affinity than the reported one.

The DM interacting residues of VSTx1, which were identified by the CS experiment, form the hydrophobic patch on the molecular surface of VSTx1, whereas the VSD binding residues of VSTx1 (Val
Figure 5. ASCS experiments to identify intermolecular proximal residue pairs. (a–c) Phe-, Ile- and Leu-selective CS experiments. The difference in the intensity reduction ratios (ΔRR) in the presence of Phe, Ile or Leu selectively 1H-labeled in a 2H-background and uniformly 2H-labeled VSD were plotted against the residue numbers of VSTx1, respectively (left). The residues indicated by asterisks are those with the signal-to-noise ratios less than 10. The error bars were calculated based on the signal-to-noise ratios. Bars corresponding to the residues with significant intensity reductions (minimum values of ΔRR > 0.1) are colored. The backbone amide hydrogen atoms of the affected residues are shown as balls in the VSTx1 structure (PDB code 1S6X) and the candidates of the CS-source residues are shown as sticks in the VSD structure (PDB code 2KYH) (right). (d) Proximal residue pairs connected by dotted lines.

Figure 6. Structural model of the VSTx1-VSD complex satisfying the ASCS results. VSTx1 and VSD are depicted by a cyan tube and a white ribbon diagrams, respectively. Their binding interface is enlarged, in which the most proximal hydrogen atoms reflecting the ASCS results are connected by dotted lines.
provide the inhibitory mechanism of KvAP by VSTx1 (Fig. 8). Under the resting membrane potential, 
and intact KvAP39. The present results could be indirect, and the binding residues of VSTx1 identified here could be occluded due to the difference in the structure and orientation to the pore domain of VSD between the K2.1 chimera channel and intact KAP28.

The VSTx1 residues in the hydrophobic patch, which interact with DM, are exposed in the complex with VSD and thus the DM molecules would surround the complex as illustrated in Fig. 7a, where the polar residues of VSTx1 in the periphery of the hydrophobic patch lie in the extracellular side (Fig. 7b left). Therefore, in the lipid bilayer, the hydrophobic patch and peripheral polar residues of VSTx1 would contribute to the interactions with acyl and polar head groups of phospholipid molecules (Fig. 7b right), which is consistent with the notion that VSTx1 partitions at the outer leaflet of the membrane40. This schematic view in the membrane is similar to the structural model from the multiscale molecular dynamics (MD) simulations using palmitoyl-oleoyl phosphatidylcholine (POPC) membrane40. By comparing the residues of VSD that were identified in the ASCS experiments with those that showed significant toxin/VSD contacts during the MD simulations, two consensus residues in S4 (F124 and L125) are indicated, but the orientation of VSTx1 to S4 is slightly different and the C-terminus of S1 and the S1–S2 linker of VSD were also involved in the binding site of VSTx1 in the MD simulations. Although this difference might be due to the surrounding environment (DM versus POPC molecules) of VSD, both interaction modes would be physiologically acceptable, since VSTx1 partitioning into the membrane27,21,23,25 could bind to VSD from different directions. Each structural model might capture one of the most stable conformations in the presence of the DM and POPC molecules, respectively.

Figure 7c shows our structure of the VSTx1-VSD complex superimposed onto the K1.2–K2.1 chimera45, indicating that there is apparently no room around the S4 of VSD for the binding of gating modifier toxins since S4 faces to the pore domain (Fig. 7c left). However, previous EPR data indicated that the arrangement of VSD and pore domain in KAP is different from that of K1.2–K2.1 chimera, and that VSD in KAP interacts with pore domain via S1 and S229. Thus, in the structure of KAP, the VSTx1 binding surface on S4 seems surrounded by lipid molecules in the absence of VSTx1, and VSTx1 can bind to the S4 of VSD with the hydrophobic patch facing to the lipid side without steric hindrance with the pore domain (Fig. 7c right).

Under the resting membrane potential, VSD adopts a down conformation, which keeps the channel closed. Upon depolarization, K+ is activated by changing the VSD conformation from down to up, and then spontaneously inactivated by decreasing population of K+-permeable state and increasing that of impermeable state42–44. Upon repolarization, K+-returns to the resting conformation. The present results provide the inhibitory mechanism of KAP by VSTx1 (Fig. 8). Under the resting membrane potential, VSTx1 is supposed to sit on the outer leaflet of the cell membrane with the hydrophobic patch buried in the membrane. Upon depolarization, where VSD adopts an up conformation with a part of the S4 helix exposed to extracellular side, the extracellular part of S4 is recognized by VSTx1. By stabilizing the up conformation of VSD, VSTx1 would delay its return to the down conformation upon repolarization and thus prolong the inactivation of KAP. This inhibitory mechanism is consistent with the recent electro-physiological data28. Actually, the binding residues and mechanism of action of the gating modifier toxins, as well as the key elements in the VSD are inconsistent in several reports11,18,20,22,26,45–47, and it would be reasonable to consider that the gating modifier toxins differ in their binding surfaces, critical residues on VSD, and inhibitory mechanisms individually. Our results, however, provide important insights into one of the mechanisms of action of the gating modifier toxins on K+ channels.

**Methods**

**Expression and purification of VSTx1.** The DNA encoding VSTx1 was inserted into the pET-30 Xa/LIC plasmid (Novagen) with a Factor Xa cleavage site between an N-terminal hexahistidine tag and VSTx1 sequences. VSTx1 was expressed in *E. coli* BL21 (DE3) cells. The uniformly 15N- and 13C, 15N-labeled samples were prepared by growing the transformants in M9 minimal medium containing 15NH4Cl, glucose or [13C]glucose and [15N] or [13C, 15N] Celto® base powder, respectively. Similarly, the uniformly 2H, 15N-labeled samples were prepared by growing the transformants in M9 minimal medium containing 99% 2H2O, 15NH4Cl, [2H]glucose and [2H, 15N] Celto® base powder. The cells were grown at 37°C to an A600 of 0.6 to 0.8 and protein expression was induced with 1.0 mM isopropyl β-D-1-thiogalactopyranoside (IPTG) for 6 h at 37°C. The cells were harvested and lysed by sonication, followed by centrifugation. The pellets were solubilized by 8.0 M urea. VSTx1 was purified from the
supernatant of the centrifuged lysate by a HIS-Select® Nickel Affinity Gel (Sigma) column. The eluted VSTx1 was refolded by dialysis, followed by digestion with Factor Xa protease (Novagen). The cleaved hexahistidine tags were removed by HIS-Select® Nickel Affinity Gel column. VSTx1 was further purified by the reverse phase high performance liquid chromatography using an ODS-AM column (YMC).

**Expression and purification of VSD.** The DNA encoding Met -12 to Lys 136 of KAP channel with an N-terminal decahistidine tag followed by a HRV 3C protease cleavage site was inserted into the pMAL-c2X plasmid (New England Biolabs). For preventing the artificial dimerization of VSD, Cys -2 was substituted to Ser by the QuikChange® system (Stratagene). Furthermore, site-specific cysteine mutations described below were also introduced by QuikChange®. VSD was expressed in _E. coli_ C41
(DE3) cells with an N-terminal maltose binding protein (MBP) and a decahistidine tag. The uniformly
$^2$H- and $^2$H, $^{15}$N-labeled samples were prepared by growing the transformants in M9 minimal medium
containing 99% $^2$H$_2$O, $[^2$H$]$ glucose, NH$_4$Cl or $[^2$H$]$NH$_4$Cl and $[^2$H$]$ or $[^2$H, $^{15}$N$]$ Celtone® base powder.
The cells were grown at 37 °C to an $A_{600}$ of 0.6 to 0.8 and protein expression was induced with 1.0 mM
IPTG for 6 h at 37 °C. For the expression of the Phe, Ile and Leu selectively $^1$H-labeled samples, a 5-fold
concentration of the target $^1$H-amino acid as compared to that of the $^2$H-amino acid included in $[^2$H$]$ Celtone® base powder was added at the same time of the addition of IPTG. The cells were harvested
and lysed by sonication, followed by centrifugation. The pellets were solubilized by DM and centrifuged.
VSD was then purified from the supernatant of the centrifuged lysate by HIS-Select® Nickel Affinity Gel
(Sigma) column and Amylose Resin (New England Biolabs) column, followed by digestion with HRV
3C protease (Novagen). The cleared MBP and decahistidine tags were removed by HIS-Select® Nickel
Affinity Gel column. The gel filtration analysis using a Superdex 200 10/300 GL column (GE Healthcare)
indicated that the purified VSD in DM was a monomer.

Gel shift assay of VSD Cys mutants by Mal-PEG 2k. VSD Cys mutants were reduced with 5.0 mM
DTT and desalted on a Micro Bio-Spin® column (Bio-Rad). The reactions were performed in a buffer
containing 50 mM Tris-HCl (pH7.0), 100 mM KCl and 4 mM DM with Mal-PEG 2k (NOF) at a molar
ratio of 1:20 for each protein:Mal-PEG at room temperature for 10 min. The excess Mal-PEG was removed
by a Micro Bio-Spin® column before non-reducing SDS-PAGE.

Fluorescence assay of VSD in POPE/POPG liposome. V119C mutant of VSD was reduced with
5.0 mM DTT and desalted on a Micro Bio-Spin® column. The reaction was performed in a buffer con-
taining 50 mM Tris-HCl (pH 7.2) and 4 mM DM with monobromobimane (mBBr, Invitrogen) at a molar
ratio of 1:10 for protein:mBBr at room temperature for 1 h. The excess mBBr was removed by a Micro
Bio-Spin® column.

A 3:1 mixture of POPE/POPG (1-palmitoyl-2-oleoyl-sn-glycero-3-phosphoethanolamine/1-palmitoyl-2-o-
leoyl-sn-glycero-3-phospho-(1′-rac-glycerol), Avanti) was dried under nitrogen stream and then resus-
pended to 10 mg/mL in dialysis buffer containing 20 mM Hepes-KOH (pH 7.0), 150 mM KCl, 10% glycerol
and 0.2 mM ethylenediamine tetraacetic acid (EDTA). The lipid mixture was sonicated in a bath sonicator.
DM was added to the lipid mixture to 20 mM and the mixture was rotated at room temperature for 1 h.
mBBr-labeled VSD was added to the lipid mixture at a weight ratio of 1:200 for protein:lipid, rotated at
room temperature for 2 h and then dialyzed against dialysis buffer for 3 days at 4 °C, exchanging buffer twice
a day. The liposomal solution was extruded on Avanti® Mini-Extruder (Avanti) with 11 passes through a
0.1 μm Nuclepore membrane (Whatman).

The resulting proteoliposomes were diluted 20-fold into a buffer containing 20 mM Hepes-KOH (pH
7.0), 150 mM NaCl, 10% glycerol, 0.2 mM EDTA and 47.5, 15, 4.75, 1.5 or 0.475 mM KCl, resulting
in the theoretical membrane potentials of 30, 60, 90, 120 or 150 mV upon addition of valinomycin (Calbiochem), respectively. Data were collected at 25 °C on a RF-5300PC spectrofluorometer (Shimadzu)
in time-acquisition mode at 1 s intervals with excitation at 394 nm, emission at 470 nm and bandwidth of
5 nm. A baseline was collected for 60 s at 0 mV and then 20 nM valinomycin was added to produce the
membrane potential. The increase in the fluorescence intensities immediately after the addition of valin-
omyacin was extrapolated from the exponential fittings of the intensity decays of each experiment. For the
measurement of the inhibition of the voltagedependent conformational change of VSD, data were col-
lected as described above after incubation with an excess amount of VSTx1 at room temperature for 1 h.

Isothermal titration calorimetry. Binding of VSTx1 to VSD in DM micelles was characterized by
isothermal titration calorimetry (ITC) using an ITC 200 MicroCalorimeter (MicroCal). VSD was placed
into a dialysis membrane (molecular weight cutoff of 50 K) and dialyzed against a buffer containing
10 mM PIPES-NaOH (pH 6.5) and 4 mM DM at 4 °C for 1 day. VSTx1 was lyophilized and diluted into

Figure 8. Schematic representation of the inhibitory mechanism of KAP by VSTx1. KAP is illustrated
by a pore domain accompanied by two S4, in which voltage sensing Arg residues are depicted by “+”. VSD
and DM binding sites on VSTx1 are colored magenta and cyan, respectively.
the external solution used for the dialysis of VSD. 36μM VSD was titrated with 350μM VSTx1 at 25°C. Heats of dilution were determined by titration into the external buffer of dialysis and subtracted from the raw titration data before analysis using the MicroCal Origin software version 5.0 provided by the manufacture. A single-site binding model was assumed. The thermodynamic parameters with error values were calculated from the fitting.

NMR analyses. NMR samples were prepared in a buffer containing 10 mM Bis-Tris (pH 6.5), 95% H₂O and 5% D₂O whereas a buffer containing 30% H₂O and 70% D₂O was used in the CS and ASCS experiments. NMR spectra were recorded at 25°C and 45°C on a Bruker Avance 600 spectrometer equipped with a triple axis gradient probe and a Bruker Avance 800 spectrometer equipped with a cryogenic probe. NMR data processing and analysis were performed using Topspin 2.1 (Bruker) and Sparky (T. D. Goddard and D. G. Kneller, Sparky 3, University of California, San Francisco). The error bars were based on the signal-to-noise ratio calculated by the Sparky software.

Sequential assignments of the backbone resonances of VSTx1 were established by HNCACB, CBCA(CO)NH and C(CO)NH experiments at 45°C. Titration of uniformly 15N-labeled VSTx1 with DM or unlabeled VSD were monitored by 1H-15N HSQC spectra, whereas titrations of uniformly 1H, 15N-labeled VSD with unlabeled VSTx1 were monitored by 1H-15N TROSY spectra. The CS and ASCS experiments were performed at 25°C on a Bruker Avance 800 spectrometer using a mixture of 200μM uniformly 1H, 15N-labeled VSTx1 and 400μM differently labeled VSD in 50 mM DM. For the conventional CS and its negative control experiments, unlabeled and uniformly 1H-labeled VSD were used, respectively. The ASCS experiments were performed using the Phe, Ile and Leu selectively 1H-labeled VSD, respectively. Irradiation was carried out for 1.0 or 1.5 s by a WURST-2 decoupling scheme. The saturation frequency was set at 1.5 ppm and the maximum radiofrequency amplitude was set to 0.17 kHz for WURST-2. The recycling delay was set at 4.0 s. In the CS and ASCS experiments, Lys 4, Ser 12, Ser 23, Arg 24, Lys 26, Leu 30 and Ala 31 were out of analysis due to their low signal-to-noise ratios less than 10.

The CS-source residues on VSD were identified by systematically combining the ASCS results based on the spatial complementarity of the cross-saturated amide hydrogen atoms on VSTx1 and the CS-source candidates on VSD. A conformer of each of the solution structures of VSTx1 (PDB code 1S6X) and VSD (PDB code 2KYH) were used for the analysis. The coordinates of the hydrogen atoms were generated using MOLMOL 2k.2.

Identification of the proximal residue pairs of the VSTx1-VSD complex. All possible combinations of the CS-source residues were systematically evaluated based on the maximum deviation of the CS-source residues and the cross-saturated amide hydrogen atoms31. The top 20 candidates of the proximal residue pairs of VSTx1 and VSD are listed according to the maximum deviations of the amide hydrogen atoms of VSTx1 and each candidate residue on VSD (Supplementary Table S2). It should be noted that the maximum deviations are not the real distances between hydrogen atoms, where the coordinates of the CS-donor residues are replaced by the center of gravity of the hydrogen atoms. The candidates proposed by the ASCS results are clustered into 3 groups. Of these groups, group 2 and 3 are not adoptable as the candidates for the proximal residue pairs due to their steric hindrances between V20 of VSTx1 and T130 on VSD (group 2) and W25 of VSTx1 and L125 or L128 on VSD (group 3), respectively. For the subsequent docking, the candidate with the smallest maximum deviation in group 1 (denoted by an asterisk in Supplementary Table S2) was utilized as the distance constraints.

Construction of the structural model of the VSTx1-VSD complex. The structural model of VSTx1-VSD complex was generated with the HADDOCK software37. The distance restraints file was generated as the unambiguous distance restraints so that the cross-saturated amide hydrogen atoms 31. The top 20 candidates of the proximal residue pairs of VSTx1 and VSD were monitored by 1H-15N TROSY spectra. The CS and ASCS experiments were performed at 25°C on a Bruker Avance 800 spectrometer equipped with a cryogenic probe. NMR data processing and analysis were performed using Topspin 2.1 (Bruker) and Sparky (T. D. Goddard and D. G. Kneller, Sparky 3, University of California, San Francisco). The error bars were based on the signal-to-noise ratio calculated by the Sparky software.

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Author Contributions
S.O., I.S. and M.O. designed the study, S.O., T.K., T.N. and H.H. performed the experiments, and S.O., I.S. and M.O. analyzed the data and wrote the paper.

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