Concentration of Heavy Metals in Borehole Water from Ikono Urban, Ikono Local Government Area, Akwa Ibom State, Nigeria

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Abstract: The concentration level of heavy metals (Pb, As, Cu and Fe) were assessed in five borehole waters from Ikono Urban. Atomic Absorption Spectrometer was used to analyze the water samples. The results showed that Pb ranged from 0.15±0.040 - 0.40±0.094mg/L; As, 0.05±0.008-0.10±0.041mg/L; Cu, 0.02±0.008-0.06±0.016mg/L and Fe, 0.20±0.016 -0.37±0.025mg/L. The concentration levels of Pb and As in the borehole samples were above WHO, NAFDAC and SON permissible/allowable limit, Cu is lower than the allowable limit while Fe is within the permissible limit for drinking water. The contamination factor calculated ranged from excessively polluted for Pb and As, very severely contaminated for Fe and very slightly contaminated for Cu. The pollution index revealed that the five boreholes waters were slightly polluted. The study showed that the water from the boreholes in Ikono Urban were contaminated with heavy metals and it has a far reaching effect on human health and therefore should be treated before being used for drinking purposes.

Keywords: Borehole, heavy metals, pollution, contamination, groundwater, water samples

1. INTRODUCTION

The activities of man have resulted in the contamination and pollution of nature. As a result of these continuous activities man’s natural environment such as soil (land), water and air have been greatly degraded. The continuous neglect and improper implementation of programmes to mitigate these activities may have resultant effect on the life of man and other organisms that occupy the face of the earth (Adesomoye et al., 2006; Edori and Kpee, 2016). For life to effectively function, water plays an important role. The ability of man to access portable water for drinking is purely based on its availability. The wellness of life, be it man or other living creatures, depends on the availability of water for drinking and other purposes (Halilu et al., 2011). The proper use of water resources and its reach to man is a worldwide challenge, most especially in the continents of Africa and Asia (WHO, 2004).

Water as an essential tool for life on the surface of the earth is used in domestic, industrial and agricultural purposes. As the population of the world increases, there is the need for the supply of safe water for drinking, domestic, industrial and agricultural uses, so that healthy life can be achieved (Elinge et al., 2012). Water plays significant roles in the growing, development and establishment of cities and communities (Waziri et al., 2009), hence man rely on water for proper existence.

In Nigeria, due to the inability of the various levels of government to provide water for its citizens, there has arisen private boreholes indiscriminately drilled by individuals, corporate organizations and even government agencies in their different homes, and office environments (LAWMA, 2000; Edori and Kpee, 2016) to curb the menace of inadequate water supply. The alternative sources of water supply provided is uncontrolled and therefore produces negative implications on the groundwater and surface water which negates the principle of sustainable development and its goals (Abii and Nwabienvanne, 2013).

Water being a universal solvent makes it vulnerable to be easily contaminated. Polluted water is not suitable for use in all ramifications. Human influence and activities such as metal processing, oil exploration and exploitation mining and agriculture have contributed to water pollution (Kolo and...
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Baba, 2004; Adeyemi and Awokunmi, 2010; Edorietal., 2016). In general, water can be primarily obtained from surface water and underground water. Underground water includes well water, borehole water while surface water sources are rivers, lakes, oceans, estuaries, creeks, streams, etc. (McMurry and Fay, 2004).

Principally, most natural water sources for agriculture and domestic uses are got underground (boreholes) (Belkhiri etal., 2018). Presently the activities of man due to advancement and urbanization has resulted in contaminating groundwater (Ozturk, etal. 2009; Momodu and Anyakora, 2010). As population increases, the desire to use land also increases and this has resulted in the pollution and contamination of the groundwater due to its vulnerability. Anthropogenic behaviours can discharge wastes and chemicals into the environment either accidentally or intentionally, which has brought about the pollution and contamination of groundwater. The remediation of contaminated groundwater is difficult and of great financial implications to restore to its normal state (Belkhiri etal., 2018).

The contamination of groundwater (boreholes) introduces heavy metals into it, although heavy metals also exist naturally underground. Heavy metals are well known contaminants of underground water, and they constitute one of the major pollutants affecting the underground water (boreholes) system (Marcovechioetal. 2007). Heavy metals are sometimes referred to as trace metals and have high density and tends to accumulate in any system not properly checked (Elingeetal., 2011). Heavy metals may be toxic and their toxicity depends on the amount available in the environment. Heavy metals leach to the groundwater when the capacity of the surface soil fails to retain them and through consumption may pass to man and other living organisms. Heavy metals are indestructible (Underwood, 1956, Edorietal., 2016). Heavy metals have the ability to exist in particulate, dissolved and colloidal phases when found in water (Adepoju-Bello et al, 2009).

The aim of this work is to determine the level of contamination of borehole waters in Ikono Urban, Ikono Local Government Area of AkwaIbom State, Nigeria.

2. MATERIALS AND METHODS

Water samples were collected from five locations within the Ikono Urban in the Ikono Local Government Area of AkwaIbom State in a one litrejerrycan. The jerrycans were first rinsed with distilled water and then with the borehole water before filling it with the water for analysis. A representative sample was formed after three water samples were collected from each borehole at intervals of four hours, between the hours of 7.00 am to 3.00 pm. The samples were collected in the months of December, February and April. Samples from the different stations were labelled L1, L2, L3, L4 and L5 respectively. On the same day of sampling, the samples were transported to the Jaros Base Laboratory, Port Harcourt, and Rivers State for heavy metal analysis.

Atomic absorption Spectrophotometer (AAS) was used for the determination of the concentration of heavy metals in the borehole water samples. The results of the concentration of the heavy metals obtained were presented in mg/L. factor (CF) and pollution index (PI) values from Lacatusu (2000).

The contamination factor (CF) is mathematically stated as,

\[ Cm/Cb \]

Where,

- \( Cm \) = metal concentration
- \( Cb \) = base/background value

The background value for this work was taken from WHO (2004) limit for drinking water.

The pollution index (PI) is mathematically stated as,

\[ PI = (CF1 \times CF2 \times CF3 \times CF4 \ldots \times CFn)^{1/4} \]

This index is used in assessing the level of pollution of heavy metals in the environment (Usero etal., 2000). Where;

- CF = contamination factor of the individual metals and n = number of metals involved.

The purpose of the contamination factor and the pollution index was to measure the degree of pollution/contamination of the boreholes by heavy metals. When the contamination/pollution index value is less than one it shows contamination range while when it is greater than one it shows...
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One of the most essential and invaluable natural resources known to man is water. Water is very essential in the support for life of man on the earth, and its availability to man is a necessity for human welfare (Kumar, 2004; Ademanya, et al., 2013; Edoriet al., 2016). Water of good quality is indispensable and its availability to man helps in preventing water borne diseases and helps in improving man’s quality of life (Ohuduro and Anyakora, 2007). As a result of man’s activities or other natural factors, water comes into contact with heavy metals which are also naturally occurring. These heavy metals dissolve in the water and contaminate it. These heavy metals at a certain level of concentration in the water becomes toxic and harmful to man and other creatures that consume it (Haliluet al., 2011). The contamination of groundwater by heavy metals are of interest today because, some even at low concentrations may be toxic to humans and animals (Marcovecchio et al., 2007). However, some heavy metals when in trace quantity have significant roles in man’s biochemistry, physiology and metabolism (Vinodhini and Narayanan, 2009).

Lead (Pb) is among the non-essential heavy metals. The mean quantity of lead recorded in the various borehole samples of Ikono Urban were above the required limit for drinking water by WHO (2004), SON (2007) and NAFDAC (2011). The presence of lead in underground water (borehole) may be as a result of corrosion of plumbing works in households, industrial and mining activities and erosion (EPA, 2005), and can also be due to exploitation and exploration of crude oil and in the use of fuel laded with lead (Kpee and Edori, 2014). Lead (Pb) when present at a high concentration results in health problems to man. Hemoglobin synthesis is inhibited by lead, and due to its carcinogenic properties can cause damage to the liver and bring about hair loss in man. When man is constantly exposed to lead (Pb), there is the possibility of delay in mental and physical development in infants and children, while in adults can cause kidney problems and high blood pressure (Imam et al., 2018). During the developmental stages of the foetus in pregnant women, the presence of lead at certain concentration can affect it and bring about disorder, even infants are vulnerable to disorder as a result of lead presence at high concentrations (Ubong and Gobo, 2001; Iyamaeta l., 2014). High concentration of lead in the blood causes brain damage, irritation and also a disease known as plumbism (Edori and Edori, 2012).

Arsenic as a heavy metal prefers to exist in water at the oxidation state of +5 and at the oxidation state of +3 becomes more stable because of the absence of oxygen (WHO, 2010). The presence of arsenic in the waters of the five boreholes of Ikono Urban were all above the WHO, and other standard organizations acceptable limits for drinking water. Aarsenic when present in drinking water constitute health challenges to man and animals, and even at very low concentrations exposure to it may prove poisonous (UNICEF, 2008). One of the most environmental poisonous contaminants known is arsenic. In both oxidation states (+5 and +3) it can be easily absorbed by the gastrointestinal tract (Smith and Steinmaus, 2007). The toxicity of arsenic is manifested in a disease known as arsenicosis, which can take about 2-20 years to appear because of its slow nature of occurrence. Toxicity due to arsenic will bring about numerous signs such as; hypo and hyper pigmentation, peripheral vascular disease, peripheral neuropathy etc. children are more vulnerable to arsenic than adults, it inhibits
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intellectual development in children, stigmatization due to skin lesion that resulted from arsenicosis. This may eventually ruin the lives of growing individuals and even their families at large (UNICEF, 2013).

The mean value of copper observed from the different borehole water samples in Ikono Urban were all below the WHO, SON and NAFDAC acceptable limits allowed for drinking water. Copper is one of the essential metals needed by man for life sustenance. Copper is important in the formation of foetus, building man’s immune system, development of the brain, the transmission of message by the neurons and anti-oxidative properties (Edori and Kpee, 2016). However, at high concentrations in the human body can lead to kidney and liver problems and eventual damage (EPA, 2005), irritation of the intestine and stomach, and over exposure to copper can lead to greater health hazard and risk especially for people with Wilson disease (Spinazzietal., 2007; Edorietal., 2016).

The concentration of the samples of the borehole waters from the five locations fall within the acceptable limit allowable for drinking water by the different standard organizations referenced in this work. The presence of iron in underground water at high concentrations may be due to anthropogenic activities such as welding and waste generated by man. Iron being one of the essential metals helps in the synthesis of haem proteins and the red colouration in blood. The human system also need iron to fight disease causing bacteria (Adriano, 2001). The concentration of iron above allowable level can result in cell damage of the gastrointestinal tract, heart and liver. Iron in excess may result in metabolic and genetic diseases which bring about blood transfusion repeatedly in patients (Anakeetal., 2014; Iyamaetal; 2014). The brown colouration in water comes about when iron is in excess and then exposed to the air, which then converts the iron from its oxidation state of +2 to +3.

Table1. Concentrations of Heavy Metals in Borehole Samples from Ikono Urban in AkwaIbom State in December

| Heavy Metals (mg/L) | Stations          | NdinyaMfia (L1) | NungUkim (L2) | ItonOdoro (L3) | IbiakuNtokOkpo (L4) | NdiyaEtok (L5) |
|---------------------|-------------------|-----------------|---------------|----------------|---------------------|----------------|
| Pb                  |                   | 0.2             | 0.5           | 0.3            | 0.5                 | 0.3            |
| As                  |                   | 0.12            | 0.08          | 0.10           | 0.14                | 0.06           |
| Cu                  |                   | 0.05            | 0.03          | 0.06           | 0.02                | 0.04           |
| Fe                  |                   | 0.26            | 0.31          | 0.29           | 0.18                | 0.40           |

Table2. Concentrations of Heavy Metals in Borehole Samples from Ikono Urban in AkwaIbom State in February

| Heavy Metals (mg/L) | Stations          | NdinyaMfia (L1) | NungUkim (L2) | ItonOdoro (L3) | IbiakuNtokOkpo (L4) | NdiyaEtok (L5) |
|---------------------|-------------------|-----------------|---------------|----------------|---------------------|----------------|
| Pb                  |                   | 0.1             | 0.2           | 0.1            | 0.3                 | 0.24           |
| As                  |                   | 0.08            | 0.08          | 0.06           | 0.06                | 0.04           |
| Cu                  |                   | 0.04            | 0.02          | 0.04           | 0.01                | 0.03           |
| Fe                  |                   | 0.29            | 0.32          | 0.36           | 0.20                | 0.34           |

Table3. Concentrations of Heavy Metals in Borehole Samples from Ikono Urban in AkwaIbom State in April

| Heavy Metals (mg/L) | Stations          | NdinyaMfia (L1) | NungUkim (L2) | ItonOdoro (L3) | IbiakuNtokOkpo (L4) | NdiyaEtok (L5) |
|---------------------|-------------------|-----------------|---------------|----------------|---------------------|----------------|
| Pb                  |                   | 0.15            | 0.20          | 0.20           | 0.50                | 0.27           |
| As                  |                   | 0.18            | 0.08          | 0.08           | 0.10                | 0.05           |
| Cu                  |                   | 0.06            | 0.04          | 0.08           | 0.03                | 0.05           |
| Fe                  |                   | 0.27            | 0.33          | 0.33           | 0.22                | 0.38           |

Table4. Mean Concentrations of Heavy Metals in Borehole Samples from Ikono Urban in AkwaIbom State in the Sampled Months

| Heavy Metals (mg/L) | Stations          | NdinyaMfia (L1) | NungUkim (L2) | ItonOdoro (L3) | IbiakuNtokOkpo (L4) | NdiyaEtok (L5) |
|---------------------|-------------------|-----------------|---------------|----------------|---------------------|----------------|
| Pb                  |                   | 0.15±0.040      | 0.30±0.141    | 0.20±0.082     | 0.40±0.094          | 0.27±0.025     |
| As                  |                   | 0.10±0.041      | 0.08±0.00     | 0.08±0.016     | 0.10±0.033          | 0.05±0.008     |
| Cu                  |                   | 0.05±0.008      | 0.03±0.008    | 0.06±0.016     | 0.02±0.008          | 0.04±0.008     |
| Fe                  |                   | 0.27±0.012      | 0.03±0.008    | 0.33±0.029     | 0.20±0.016          | 0.37±0.025     |

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**Table 5. Approved Standards for the Selected Metals in Drinking Water**

| Heavy Metals | NAFDAC | SON | WHO   |
|--------------|--------|-----|-------|
| Pb           | 0.01   | 0.01| 0.01  |
| As           | 0.01   | 0.01| 0.01  |
| Cu           | 1.00   | 1.00| 0.5–2.0 |
| Fe           | 0.30   | 0.30| 0.30  |

The level of contamination and pollution of the boreholes when observed from the contamination factor and pollution index significant intervals (Lacatusu, 2000), a standard normally used to verify the level of contamination/pollution due to contamination effect from heavy metals revealed that the borehole waters are excessively polluted with Pb and As, Cu is very slightly contaminated and Fe is very severely contaminated. Considering the heavy metals used (Pb, As, Cu and Fe), the pollution index from the results showed that water from the different boreholes were slightly polluted.

The contamination/pollution index values according to Lacatusu (2000) are; < 0.1 is very slightly contaminated, 0.10 - 0.25 is slightly contaminated, 0.26 – 0.5 is moderately contaminated, 0.51 – 0.75 is severely contaminated, 0.76 – 1.00 is very severely contaminated, 1.1 – 2.0 is slightly polluted, 2.1 – 4.0 is moderately polluted, 4.1 – 8.0 is severely polluted, 8.1 – 16.0 is very severely polluted and > 16 is excessively polluted.

**Table 6. Contamination Factor and Pollution Index of Borehole Water from Ikono Urban in AkwaIbom State**

| Heavy Metals | NdinyaMfia (L1) | NungUkim (L2) | ItonOdoro (L3) | IbiakuNtokOkpo (L4) | NdiyaEtok (L5) |
|--------------|----------------|---------------|----------------|---------------------|----------------|
| Pb           | 15             | 30            | 20             | 40                  | 27             |
| As           | 10             | 8             | 8              | 10                  | 5              |
| Cu           | 0.025          | 0.015         | 0.03           | 0.01                | 0.02           |
| Fe           | 0.9            | 1.07          | 1.10           | 0.67                | 1.23           |
| Pollution Index | 1.36          | 1.40          | 1.52           | 1.28                | 1.35           |

4. **CONCLUSION**

The study revealed that the concentration level of heavy metals in the different boreholes within the Ikono Urban is at a level that needed to be controlled, especially Pb and as which were above the WHO allowable limit for drinking water. The poor quality of drinking water within the locality is typical of a third world country like Nigeria. Urgent steps should be taken to forestall accumulation in order to overcome any water borne disease and other related health hazard due to heavy metals accumulation.

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