3D-to-2D Morphology Manipulation of Sputter-Deposited Nanoscale Silver Films on Weakly Interacting Substrates via Selective Nitrogen Deployment for Multifunctional Metal Contacts

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Cite This: ACS Appl. Nano Mater. 2020, 3, 4728−4738

ABSTRACT: The ability to reverse the inherent tendency of noble metals to grow in an uncontrolled three-dimensional (3D) fashion on weakly interacting substrates, including two-dimensional (2D) materials and oxides, is essential for the fabrication of high-quality multifunctional metal contacts in key enabling devices. In this study, we show that this can be effectively achieved by deploying nitrogen (N₂) gas with high temporal precision during magnetron sputtering of nanoscale silver (Ag) islands and layers on silicon dioxide (SiO₂) substrates. We employ real-time in situ film growth monitoring using spectroscopic ellipsometry, along with optical modeling in the framework of the finite-difference time-domain method, and establish that localized surface plasmon resonance (LSPR) from nanoscale Ag islands can be used to gauge the evolution of surface morphology of discontinuous layers up to a SiO₂ substrate area coverage of ∼70%. Such analysis, in combination with data on the evolution of room-temperature resistivity of electrically conductive layers, reveals that presence of N₂ in the sputtering gas atmosphere throughout all film-formation stages: (i) promotes 2D growth and smooth film surfaces and (ii) leads to an increase of the continuous-layer electrical resistivity by ∼30% compared to Ag films grown in a pure argon (Ar) ambient atmosphere. Detailed ex situ nanoscale structural analyses suggest that N₂ favors 2D morphology by suppressing island coalescence rates during initial growth stages, while it causes interruption of local epitaxial growth on Ag crystals. Using these insights, we deposit Ag layers by deploying N₂ selectively, either during the early precoalescence growth stages or after coalescence completion. We show that early N₂ deployment leads to 2D morphology without affecting the Ag-layer resistivity, while postcoalescence introduction of N₂ in the gas atmosphere further promotes formation of three-dimensional (3D) nanostructures and roughness at the film growth front. In a broader context this study generates knowledge that is relevant for the development of (i) single-step growth manipulation strategies based on selective deployment of surfactant species and (ii) real-time methodologies for tracking film and nanostructure morphological evolution using LSPR.

KEYWORDS: silver, nitrogen, thin film, growth manipulation, FDTD calculations, in situ growth monitoring

1. INTRODUCTION
Vapor-based growth of thin noble-metal films with two-dimensional (2D) morphology on weakly interacting 2D material and oxide substrates is essential for the fabrication of multifunctional metal contacts in a wide array of devices, including photodetectors,1,2 long-range surface plasmon resonance biosensors,3 and tunnel field-effect transistors for ultralight-weight mobile and wearable electronics.4,5 Noble-metal films typically exhibit a pronounced and uncontrolled three-dimensional (3D) morphology on weakly interacting substrates,4,6 which manifests itself by discontinuous layer formation during the initial film growth stages and rough surfaces upon formation of a continuous layer. From the viewpoint of thermodynamics, this growth behavior is explained by the fact that the adsorption energy of noble-metal atoms on weakly interacting surfaces is significantly smaller than the bulk metal binding energy,4,6 which provides the driving force for minimizing the metal/substrate contact area.

Film growth via vapor condensation is a far-from-equilibrium process, and hence, morphological evolution depends, primarily, on kinetic rates of atomic-scale structure-forming mechanisms.7−10 The kinetic pathways leading to 3D morphologies in metal-on-metal homoepitaxial systems are well established in the literature.10 This understanding has
facilitated the development of growth manipulation strategies, in which minority metal\textsuperscript{11–17} and gaseous\textsuperscript{18,19} species (also termed surfactants) are deposited simultaneously with the film-forming species, thereby modifying atomic-scale kinetics and allowing navigation between 3D and 2D morphologies. Numerous studies on surfactant effects for semiconductor epitaxial film growth\textsuperscript{20–23} and metal-on-metal heteroepitaxial systems\textsuperscript{24–26} are also available in the literature.

The mechanisms that govern morphological evolution of metal films on weakly interacting substrates are different than those in strongly interacting homo- and heteroepitaxial systems.\textsuperscript{6,27–31} As a result, established knowledge for surfactant-based growth manipulation is not directly applicable to the case of noble-metal-film deposition on, e.g., oxides and 2D materials. Despite this, empirical studies exist in which less-noble-metal surfactants and seed layers\textsuperscript{32–37} as well as gaseous surfactants (nitrogen (N\textsubscript{2}) and oxygen (O\textsubscript{2}))\textsuperscript{38–44} have been used to suppress 3D morphology of silver (Ag) and copper (Cu) films on oxide substrates. The presence of surfactants at the film growth front, however, is often accompanied by changes in other physical properties of the noble-metal layers (e.g., electronic, optical, and transport properties),\textsuperscript{35,36,39} which, if not reversed or mitigated, render surfactant-based approaches largely inapplicable for metal-contact synthesis. Hence, a fundamental understanding of the surfactant effects on various film-growth stages (i.e., island nucleation, island growth, island coalescence, and hole filling) is required for designing knowledge-based, efficient, and noninvasive growth manipulation strategies.

In the present work, we investigate the effects of N\textsubscript{2} on the growth of magnetron-sputtered nanoscale Ag islands and layers on silicon dioxide (SiO\textsubscript{2}) substrates, with the purpose of exploring the viability of surfactant-based approaches for metal-contact fabrication, as conceptualized in Scheme 1. Ag/SiO\textsubscript{2} is an archetypal weakly interacting film/substrate system which exhibits pronounced 3D morphological evolution.\textsuperscript{35–46} Furthermore, N\textsubscript{2} does not chemisorb on Ag surfaces, and no thermodynamically stable Ag–N compounds exist,\textsuperscript{49} which enables us to focus on the effect of N\textsubscript{2} on surface atomic-scale growth processes. By employing spectroscopic ellipsometry for real-time \textit{in situ} film growth monitoring, along with finite-difference time-domain (FDTD) calculations, we show that addition of N\textsubscript{2} to the sputtering gas atmosphere, throughout all film-growth stages, promotes in-plane nanoscale island growth and 2D film morphology, while it leads to an increase in the electrical resistivity of continuous layers, relative to Ag films grown in pure Ar. Moreover, we find—by means of detailed \textit{ex situ} microstructural analyses—that 2D morphology is promoted because N\textsubscript{2} suppresses the coalescence rate of nanoscale Ag islands, while the electrical resistivity increase is attributed to N\textsubscript{2}-induced grain refinement.

On the basis of the above insights highlighted in the previous paragraph, we design and implement a deposition protocol in which N\textsubscript{2} is introduced during Ag film growth with high temporal precision for influencing selected film-formation stages. We show that when N\textsubscript{2} is deployed only during island nucleation, growth, and coalescence, after which N\textsubscript{2} supply is turned off, such that deposition commences in an Ar/N\textsubscript{2} atmosphere and is completed in pure Ar atmosphere, 3D morphology is suppressed without compromising the Ag-layer electrical resistivity. In the opposite case, when deposition commences in an Ar atmosphere and N\textsubscript{2} is introduced after island coalescence has been largely completed, the tendency for 3D morphological evolution is dramatically enhanced. The overall results of this study provide the scientific foundation for designing advanced growth manipulation strategies based on selective deposition of surfactants to promote 2D growth without affecting the film physical properties. Such approaches can also be the basis of single-step room-temperature processes for enhancing 3D morphology and tuning the size of supported nanoparticles. Moreover, the combination of FDTD calculations and experiments highlights a path toward real-time methodologies for tracking metal film and nanostructure morphological evolution by using LSPR.

2. RESEARCH METHODOLOGY

2.1. Film Growth. Ag films are grown by direct-current magnetron sputtering (dcMS), at a constant current mode of 20 mA, on Si(001) substrates with an \textless~530 nm thick thermally grown SiO\textsubscript{2} overlay. No intentional heating is applied on the substrate, and all depositions are performed in a multisource, ultrahigh-vacuum (UHV) chamber with a base pressure below 10^{-10} Torr (\textsim\textless~10^{-8} Pa). The Ag target (purity 99.99 at. %; diameter 7.6 cm; thickness 6 mm) is placed 7.5 cm away from the substrate and at a 45° angle with respect to the substrate surface normal.

Ag depositions are performed in Ar and Ar/N\textsubscript{2} mixtures at a total working pressure \(p_{\text{total}}\) of 10 mTorr (1.33 Pa). For
samples grown in Ar/N$_2$ atmospheres, initial tests at various N$_2$ partial pressures $p_{N_2}$ showed that increasing $p_{N_2}$ from 0.05$p_{\text{total}}$ to 0.1$p_{\text{total}}$ affected film growth (as evidenced from in situ ellipsometry data by using the methodologies described in section 2.2), while no appreciable difference was observed for further $p_{N_2}$ increase up to 0.15$p_{\text{total}}$. Hence, on the basis of results of those initial tests, in this article we focus on experiments performed at $p_{N_2} = 0.1$p$_{\text{total}}$. It should be pointed out that additional effects on film properties for $p_{N_2}$ values well outside the range considered here (i.e., substantially larger than 0.15$p_{\text{total}}$) cannot be ruled out; see for example the studies by Lee et al.$^{46}$ and Wang et al.$^{53}$ on the growth of Cu–O films in Ar–O$_2$–N$_2$ gas mixtures, in which it was found that N$_2$ partial pressures above ~30% of the total working pressure affect profoundly the plasma chemistry and film electron transport properties.

Film morphological evolution, microstructure, and chemical composition are analyzed by using the in situ and ex situ tools and methodologies described in sections 2.2 and 2.3, respectively. Prior to exposing the samples to atmosphere for ex situ analyses, and immediately after Ag deposition, samples are capped with a 3 nm thick amorphous carbon (a-C) layer to avoid surface contamination and changes in film morphology upon atmospheric exposure. The a-C capping layers are sputter-deposited at a rate of 0.01 nm/s in the same UHV chamber from a graphite target (purity 99.99 at.%; diameter 7.6 cm; thickness 6 nm) in pure Ar discharge at 10 mTorr. The graphite target is operated with dcMS at a constant current mode of 50 mA, 7.5 cm away from the substrate, and at an angle of 45° with the substrate surface normal.

2.2. In Situ Film Growth Monitoring. Spectroscopic ellipsometry is employed to monitor the change in film optoelectronic properties during deposition and, thereby, provide real-time insights into film morphological evolution. Ellipsometric angles $\Psi$ and $\Delta$ are acquired every ~2 s at 67 incident-light photon energies in the range 1.6–3.2 eV, at an angle of incidence of ~70° from the substrate normal, by using a J.A. Woollam Inc. M-88 instrument. The acquired data are fitted to a three-phase model consisting of substrate, film, and vacuum. The substrate is modeled as a 625 μm thick Si slab with a SiO$_2$ overlayer, the thickness of which (~530 nm) is confirmed by measuring the optical response of the substrate prior to deposition. Reference data for the substrate layers are taken from Herzinger et al.$^{52}$ The optical response of the film is described by different dispersion models, depending on the film growth stage (i.e., nominal film thickness $\Theta$), as detailed below.

During initial growth stages, the film surface primarily features isolated Ag islands that support LSPR.$^{46,53}$ Being a resonant effect, LSPR can be described by adapting the Lorentz oscillator model$^{46,53}$ to express the complex dielectric function of the layer $\tilde{\varepsilon}(\omega)$ as

$$\tilde{\varepsilon}(\omega) = \frac{f_0 \omega^2}{\omega_0^2 - \omega^2 - i \Gamma \omega}$$  (1)

In eq 1, $f$ and $\omega_0$ are the oscillator strength and resonance frequency, respectively, and $\Gamma$ represents the damping rate of the plasmon resonance. It should be emphasized here that eq 1 describes the effective LSPR of the nanoscale Ag islands/air complex system, which is considered a homogeneous layer. Moreover, the single Lorentz model does not account for dipole–dipole interactions; i.e., the optical response of early growth-stage Ag layers can be accurately modeled by eq 1 as long as the distance between islands is large enough to render island–island interactions ineffective.$^{46,53}$

The optical response of electrically conductive Ag films is described by the Drude free electron theory, which is extensively used for ideal metals.$^{54}$ In this case, $\tilde{\varepsilon}_D(\omega)$ is given by the expression

$$\tilde{\varepsilon}_D(\omega) = \varepsilon_\infty - \frac{\omega_p^2}{\omega^2 + i \Gamma_p \omega}$$  (2)

In eq 2, $\varepsilon_\infty$ is a constant that accounts for the effect of interband transitions occurring at frequencies higher than the ones considered here, $\Gamma_p$ is the free-electron damping constant, and $\omega_p = \sqrt{n e^2 / \varepsilon_0 m_e}$ is the free-electron plasma frequency, where $n$ is the free-electron density, $\varepsilon_0$ is the permittivity of free space, and $m_e$ is the free-electron effective mass. From eq 2, the room-temperature film resistivity is calculated as$^{54}$

$$\rho = \frac{\Gamma_p}{\varepsilon_\infty \omega_p^2}$$  (3)

In addition to optical properties, analysis of the ellipsometric data provides the film thickness $h_f$ as a function of deposition time $t$. Based upon $h_f$ values of continuous layers, a deposition rate $F \approx 0.1$ nm/s is obtained, irrespective of the growth conditions. This allows the nominal film thickness, $\Theta = F \times t$, to be determined to be a function of the deposition time. Here, $\Theta$ is expressed in monolayers (ML), in which 1 ML corresponds to the distance between adjacent Ag (111) atomic planes (0.236 nm). We note that $h_f$ is also determined by ex situ X-ray reflectometry measurements, as detailed in section 2.3.

2.3. Ex Situ Characterization. Crystal structure, coherence length, and film texture are determined by X-ray diffractometry (XRD) in Bragg–Brentano geometry by using a Panalytical X’pert Pro diffractometer. Film thickness, density, and roughness are determined by X-ray reflectometry (XRR), performed in a Panalytical Empyrean diffractometer. For both techniques, a copper Kα source (wavelength 0.15418 nm) in line focus is used (operated with 45 kV and 40 mA), with a parallel beam mirror and parallel plate collimator in incident and diffracted/reflected beam path, respectively. The diffracted/reflected X-ray signal is processed with a X’Celerator/PIXcel-3D detector (Malvern Panalytical) operated in scanning line mode. Moreover, a nickel filter is used in XRD measurements for removing the copper Kβ radiation.

Film morphology during initial growth stages is analyzed by plan-view transmission electron microscopy (TEM) as a function of nominal film thickness $\Theta$. Plan-view specimens are prepared by depositing a-C-capped Ag films, grown in both Ar and Ar/N$_2$ atmospheres, on electron-transparent Si$_2$O$_3$ grids, stabilized by a Formvar foil. The TEM specimens are imaged in bright- and dark-field modes by using a FEI G2 TF20UT transmission electron microscope operated at 200 kV with 0.19 nm resolution. Selected area electron diffraction (SAED) patterns are also recorded to obtain information about the film crystal structure.

The microstructure and growth morphology of Ar- and Ar/N$_2$-grown continuous Ag layers are examined by cross-sectional TEM (XTEM). Specimens are prepared by mechanical polishing, followed by Ar$^+$ ion milling in a Gatan 4730
PIPS TEM; high-resolution TEM (HRTEM) and SAED data are acquired in a Jeol 2011 UHR electron microscope, operated at 200 kV, with a point resolution of 0.194 nm and a spherical aberration of 0.5 mm.

Film chemical composition is determined by X-ray photoelectron spectroscopy (XPS). XPS measurements are performed in a Kratos AXIS Ultra DLD UHV system (base pressure ~ 3 × 10^{-10} Torr), equipped with a monochromated aluminum Kα X-ray beam, a hemispherical sector analyzer, and a multichannel detector. A 20 eV pass energy, resulting in full width at half-maximum (FWHM) of <500 meV for the Ag-3d_{5/2} peak, is used to record core-level spectra. Measurements are performed as a function of the sample depth by etching the surface with a 4 keV Ar^+ ion beam. Charge-induced shifts of the binding energy are corrected with respect to the Ar-2p peak, which originates from Ar implanted in the film during the Ar^+ ion sputtering process. The Kratos Vision software and its sensitivity factor database are used for elemental analysis. Compositional depth profiles are also acquired using time-of-flight secondary ion-mass spectrometry (ToF-SIMS) in a ToF.SIMS 5 instrument (ION-TOF; Münster, Germany) with a base pressure of 7.5 × 10^{-10} Torr. A 1 keV cesium ion (Cs^+) beam produces a 400 × 400 μm^2 crater at a rate of ~0.3 nm/s. SIMS spectra are acquired while scanning a 1 pA 30 keV bismuth ion (Bi^+) beam over sample areas of 100 × 100 μm^2. SIMS spectra are recorded every 1 s.

2.4. Finite-Difference Time-Domain Calculations.

Along with the ellipsometric analyses, we perform optical calculations based on the finite-difference time-domain (FDTD) method to evaluate the LSPR positions and compare them with the experimentally determined Lorentz-oscillator energy values (see section 2.2). We use surface morphology data (i.e., island size and number density) obtained from in situ plan-view TEM measurements (see section 2.3 for details) to compute the total reflectivity of the nanoscale Ag islands in the near-infrared, visible, and ultraviolet spectral range and thereby extract the LSPR position from the maximum of the corresponding spectra reflectivity curves. More information about the use of plan-view TEM data for FDTD calculations is given in the Supporting Information.

3. RESULTS AND DISCUSSION

3.1. Film Morphological Evolution. The room-temperature resistivities of percolated Ag films grown in pure Ar and Ar/N_2 mixtures are plotted in Figure 1 as a function of the nominal thickness Θ. The data are extracted from in situ spectroscopic ellipsometry analysis, as explained in section 2.2. Both ρ vs Θ curves exhibit a sharp initial drop until they reach a steady-state value ρss. We have previously shown that the Θ value at which ρss is established marks the completion of the hole-filling process and the formation of a continuous film. The data in Figure 1 show that addition of N_2 in the growth atmosphere results in the thickness of continuous film formation Θ_cont to decrease to 51 ML from 78 ML for the Ag film grown in pure Ar. Thus, the presence of N_2 promotes 2D growth. Furthermore, Ag growth in N_2-containing atmosphere yields a steady-state resistivity ρss = 1.4 × 10^{-5} Ω-cm, which is larger than the corresponding value 1.1 × 10^{-5} Ω-cm for films grown in pure Ar.

To correlate early film growth stages with the morphological evolution trends extracted from Figure 1, we perform plan-view TEM investigations for a-C/Ag/SiO_2 samples grown in Ar and Ar/N_2-grown samples, respectively. Irrespective of the composition of the sputtering gas atmosphere, the plan-view TEM data show the expected overall film morphological evolution as a function of Θ: the initially formed isolated nanoscale islands increase in size, until they start impinging upon another, and coalesce. Further vapor deposition leads to island sizes that hinder coalescence completion, which results in the formation of an interconnected island network and eventually (for the film grown in Ar/N_2 atmosphere at Θ = 42 ML) in a nearly continuous layer.

For Θ = 2 ML, both Ar- and Ar/N_2-grown film surfaces feature nearly spherical islands with sizes in the range ~1−2 nm. Moreover, the presence of N_2 in the growth atmosphere results in a slight increase of the island number density. By analyzing data from multiple images, we find island densities of (3.97 ± 0.45) × 10^{16} and (4.13 ± 0.40) × 10^{16} m^{-2} for films deposited in Ar and Ar/N_2 atmospheres, respectively. This change by only ~5% is rather marginal, relative to multifold
increases in homoepitaxial growth experiments performed at the presence of surfactant gases.\textsuperscript{10,18} Hence, we conclude that N\textsubscript{2} at the partial pressures used in this study, does not have a significant effect on Ag adspecies surface diffusion rates during nucleation and early film-growth stages.

More pronounced differences in the film morphology are observed at larger nominal thicknesses. For \( \Theta = 8 \) ML, deposition in Ar atmosphere yields mostly spherical islands with sizes between \( \sim 2 \) and \( \sim 7 \) nm. Addition of N\textsubscript{2} leads, instead, to a film surface that hosts, predominantly, polycrystalline elongated island clusters with sizes \( \sim 5-10 \) nm. Experimental and theoretical studies on the growth of metals on weakly interacting substrates\textsuperscript{45,58} have attributed the formation of elongated cluster shapes to a decrease in the rate of island coalescence. On the basis of these studies, we conclude that the presence of N\textsubscript{2} in the gas atmosphere causes suppression of coalescence-induced cluster reshaping, promoting in-plane island growth and 2D growth morphology. In a recent study of vapor-based growth of Ag nanoparticles on ZnO in Ar/N\textsubscript{2} gas mixtures, Yun et al.\textsuperscript{44} suggested that atomic nitrogen incorporation onto nanoparticle facets lowers their surface energy and thereby the thermodynamic driving force for coalescence. This mechanism may also be relevant for explaining the results in Figure 2, but further research is required to understand the atomistic pathways by which N\textsubscript{2} suppresses the coalescence rate of nanoscale Ag islands.

At a nominal thickness \( \Theta = 21 \) ML, the plan-view TEM micrograph that corresponds to the Ar-grown sample shows the existence of isolated island clusters at various stages of coalescence, which cover \( \sim 60\% \) of the substrate surface. For the same amount of deposited material, the Ar/N\textsubscript{2}-grown sample surface consists of interconnected islands and the substrate area coverage is \( \sim 70\% \). Moreover, the presence of N\textsubscript{2} causes the appearance of smaller islands, with sizes \( \sim 1 \) nm, both in between and on the top of the elongated Ag clusters. The existence of these small islands is a signature of repeated nucleation, i.e., interruption of local epitaxial growth and nucleation of new islands with different crystallographic orientation.\textsuperscript{9,27}

An almost continuous film with only a small fraction of the substrate surface exposed is obtained at \( \Theta = 42 \) ML, when deposition is performed in an Ar/N\textsubscript{2} atmosphere. In contrast, the Ag layer is still discontinuous (substrate area coverage \( \sim 80\% \)) in pure Ar atmosphere for the same nominal thickness. The overall results presented in Figure 2 are qualitatively consistent and confirm the trends observed in Figure 1, i.e., N\textsubscript{2} addition to the growth atmosphere promotes 2D film morphological evolution.

Discontinuous Ag layers that consist of isolated islands and/or island clusters, i.e., films for \( \Theta \leq 21 \) ML according to the results presented in Figure 2, can give rise to LSPR.\textsuperscript{46,49} Using data from multiple images (like those presented in Figure 2) as input, we performed FDTD calculations and estimated the expected LSPR energies for Ar- and Ar/N\textsubscript{2}-grown films, as explained in section 2.4 and in the Supporting Information. Representative experimental (TEM) and simulated plan-view morphologies along with the corresponding calculated reflectivity spectra for \( \Theta = 2 \) ML (film deposited in Ar atmosphere) are displayed in Figures 3a, 3b, and 3c, respectively. Additional data for Ar- and Ar/N\textsubscript{2}-grown films at \( \Theta = 2 \) and 8 ML are presented in the Supporting Information. The theoretical LSPR position corresponds to the wavelength at which the reflectivity curve exhibits its maximum value, as marked by the vertical dotted line in Figure 3c.

Figure 3a presents the FDTD-predicted LSPR position (full symbols) along with the Lorentz-model resonance energy \( \hbar \omega_{0} \) (hollow symbols), obtained from \textit{in situ} ellipsometric analyses, as a function of \( \Theta \). Irrespective of the composition of the sputtering atmosphere, both calculated and experimentally obtained resonance energy values exhibit qualitatively consistent trends vs \( \Theta \) (i.e., red-shift), while they are in fairly good quantitative agreement. This agreement confirms that the
Lorentz model is an accurate description of the effective resonant response of the nanoscale Ag islands on the surface of the substrate during the initial film-growth stages.

Upon vapor deposition up to ~2 ML, the LSPR/ℏω₀ vs Θ curves for both Ar- and Ar/N₂-grown films are essentially identical, indicating no significant difference in the film morphology, which is consistent with plan-view TEM results for Θ = 2 ML in Figure 2. The systematic shift of the FDTD-calculated LSPR positions to lower values relative to the ellipsometry-derived ℏω₀ at Θ = 2 ML can be attributed to (i) the use of a correction factor to the electron scattering time in the FDTD calculations (see the Supporting Information), which may successfully capture the major effects of plasmon damping, but overestimates the plasmon blue-shift; (ii) the fact that classical electromagnetism, used in FDTD, cannot account for nonlocalities, including blue-shift and quenching of the plasmon resonance of very small particle sizes; and (iii) the low sample reflectivity at the initial growth stages (i.e., small Θ values).

For Θ > 2 ML, N₂ addition to the sputtering atmosphere leads to a steeper decline of the LSPR/ℏω₀ vs Θ curve compared to the Ar-deposited film. The red-shift in the LSPR/ℏω₀ position has been attributed to changes in island size and/or in inter-island separation; i.e., it reflects changes of surface areal coverage with continued deposition. To better understand the correlation among the data in Figure 4a and early stage film morphological evolution, we calculate (using FDTD) the LSPR positions for various periodic and close-packed hemispherical island arrays as a function of the substrate surface coverage (in the range 10 to 90%) for two scenarios: (i) the island number density (i.e., number of islands/μm²) is kept constant, and the surface coverage is varied by changing the island volume (i.e., diameter); (ii) the island diameter is kept constant, and the surface coverage is varied by changing the island number density. The results of these calculations are shown in Figure 4b, whereby the contour plot is a color-coded map which depicts the LSPR evolution for the scenarios i and ii above. The island iso-population points are indicated by black dashed lines, while the gray-line-shaded region denotes surfaces on which islands are separated by <1 nm, and other nonlocal effects make the FDTD calculations inaccurate (more details on the calculations can be found in the Supporting Information). From the data in Figure 4b, it becomes evident that an LSPR red-shift of a magnitude that is consistent with the experimental results in Figure 4a requires a significant increase of the areal coverage that is caused by a concurrent increase/decrease of island size/island–island separation. Hence, the steeper LSPR/ℏω₀ vs Θ curve slope for the Ar/N₂ grown sample (relative to the Ar-deposited one) indicates that the presence of N₂ promotes in-plane vs out-of-plane island growth, in agreement with the morphology evidenced by the plan-view TEM data for Θ = 8 and 21 ML in Figure 2. Moreover, by combining the results in Figures 1–4a, we conclude that film morphological evolution, as manifested by changes in Θ out at later growth stages, is largely set during the initial film-formation stages of island growth and coalescence.

3.2. Continuous Film Microstructure and Structure-Forming Process. The effect of adding N₂ to the sputtering atmosphere on the microstructure and chemical composition of continuous layers is also investigated. XRD analyses for ~25 nm thick Ar- and Ar/N₂-grown Ag samples are presented in Figure S4. The XRD pattern of the Ag film grown in pure Ar atmosphere shows 111 preferred orientation with a full width at half-maximum (FWHM) of 0.44°. The XRD pattern of the Ar/N₂-deposited layer exhibits no indication of strong texture formation, a much lower (111) peak intensity, and a wider FWHM of 0.88°. The latter is an indication that presence of N₂ causes the formation of a finer grain structure compared to films deposited in pure Ar.

Typical XTEM micrographs for Ar- and Ar/N₂-grown samples are shown in Figures 5a and 5b, respectively. The micrograph of the Ag layer grown in a pure Ar atmosphere (Figure 5a) reveals a microstructure composed of columnar grains with in-plane sizes up to 100 nm, which extend...
throughout the entire film growth direction. The projected film thickness, obtained from multiple positions, is found to be 27.5 ± 1.8 nm, and no discontinuities are observed, which is consistent with data in Figure 1 showing that a continuous layer is obtained at 78 ML (∼18 nm). The absence of discontinuities is also the case for the film grown in an Ar/N₂ atmosphere (Figure S5b). However, the grains of the Ar/N₂-deposited layer have more globular shapes and smaller sizes in the range 10−20 nm, and they are irregularly stacked along the growth direction. This morphology yields a similar to the Ag-grown layer mean projected thickness of 26.8 nm, but a larger statistical spread of ±3.6 nm.

The polycrystalline nature of both Ar- and Ar/N₂-grown films, evidenced by the XRD data in Figure S4, is also confirmed by the ring-type SAED patterns presented in Figure S5. Moreover, the SAED patterns—recorded under the same size selected area aperture—reveal an apparently lower number of reflections in the Ag, which indicates the occurrence of larger Ag crystals compared to the film deposited in Ar/N₂ gas mixtures. Concerning preferred growth orientation, HRTEM imaging shows that large Ag grains in Ar-grown films are (111) oriented; namely, their [111] crystal planes are normal to the [100] growth direction of the Si substrate (Figure S5c). This crystallographic texture becomes considerably less pronounced for Ar/N₂-grown films because of the random orientation of the smaller Ag crystals (Figure S5d), which is consistent with XRD analysis.

XPS analyses of Ar- and Ar/N₂-grown films, after sputter-etching ~3 nm of surface layers (see Figure S6 for spectra), show the existence of all Ag-related peaks (Ag-3s, Ag-3p, Ag-3d, Ag-4s, and Ag-4p). In addition, Ag-3d core-level high-resolution scans (also presented in Figure S6) reveal that only Ag−Ag bonds form in both films. Hence, no evidence of Ag−N compound formation is found in the XPS data, despite the presence of N₂ with a P_Ag/P_total ratio of 10% in the sputtering atmosphere. This is consistent with previous XPS studies by Depla and De Gryse. Moreover, N-1s high-resolution scans (not shown here) reveal traces of nitrogen (N) in the Ar/N₂-grown sample: on the surface of the a-C/Ag/SiO₂ stack (C=N, C≡N, and C≡N bonds) and in the vicinity of the Ag/SiO₂ interface (Si−N bonds). However, the XPS detection limit for N and other light elements is below 1 at. %, while in our Ar-grown films, the N-1s peak cannot be easily resolved, since its binding energy coincides with the Ag-3d plasmon-loss satellite peak. Hence, ToF-SIMS measurements are performed to unequivocally confirm or refute N incorporation in the Ag/N₂-grown film, as explained in the following.

The ToF-SIMS technique exhibits sensitivity in the part-per-million range, but accurate quantitative concentration determination for dopants (i.e., low content elements as N in our case) is often nontrivial due to matrix effects related to different degrees of ionization for emitted secondary ions, while the use of reference samples is required. Hence, here we focus on qualitative analysis of the film composition. In our measurements, the most significant signals in negative ion spectra stem from CN⁻ ions at 26 Da—which we relate with N and are detected in both Ar and Ar/N₂-grown layers—and Ag⁺ ions at 107 Da. We attribute the presence of N in the sample deposited in pure Ar atmosphere to contamination from atmospheric exposure of (i) the substrate prior to film growth and (ii) the sample in the time between growth and SIMS measurements. Figure 6 plots the CN⁻-to-Ag⁺ signal intensity ratio as a function of the sputtering depth (i.e., the product of the sputtering rate and the sputtering time) for Ag films grown in pure Ar (black squares) and mixed Ar/N₂ (red circles) atmospheres. The position of the film surface and the approximate position of the film/substrate interface are indicated by dashed vertical lines.

Figure 6. Ratio of CN⁻ to Ag⁺ ion signal vs sputtering depth as measured by ToF-SIMS for magnetron sputter-deposited Ag layers on SiO₂/Si substrates in pure Ar (black squares) and Ar/N₂ (red circles) atmospheres. The position of the film surface and the approximate position of the film/substrate interface are marked with blue dashed vertical lines. For both deposition conditions, the signal ratio exhibits a qualitatively similar evolution; i.e., it initially decreases when moving away from the surface into the bulk of the film and then shows a peak near the film/substrate interface. However, the CN⁻-to-Ag⁺ signal ratio obtained for the Ar/N₂-grown film is systematically larger by up to 2 orders of magnitude than that for the Ar-deposited Ag layer, in the thickness range 10−25 nm. This shows that presence of N₂ in the growth atmosphere leads to incorporation N in the sample, primarily at the Ag/SiO₂ interface as indicated by the XPS results. The absence of N in the bulk of the Ag layer may be attributed to the weak Ag−N interaction, which has been suggested to lead to N−N recombination into N₂ and desorption from the film surface.\(^{66,67}\) Accumulation of N at the film/substrate interface has also been reported by Yun et al.\(^{44}\) for Ag/ZnO heterostructures synthesized by magnetron sputtering in Ar/N₂ gas mixtures.

The occurrence of repeated nucleation in Ag films grown in mixed Ar/N₂ atmospheres is consistent with the fine-grained structure evidenced in the XRD and XTEM results in Figure S4 and Figure S5, respectively. This type of morphology can also explain the larger steady-state resistivity \(\rho\) for the Ar/N₂-deposited films shown in Figure 1; a finer-grained structure implies a larger number density of grain boundaries, which scatter charge carriers and reduce their mean-free path. Repeated nucleation has been ascribed in the literature to intense energetic bombardment during growth and/or to surfactant and impurity adsorption at the film growth front.\(^{9,27}\) In our case, combined SIMS and XPS data support temporary N adsorption during Ag film growth in Ar/N₂ atmospheres, which may lead to interruption of local epitaxy and grain refinement.\(^{9,27,67,68}\)

3.3. Growth Manipulation by Selective N₂ Deployment. The results presented in sections 3.1 and 3.2 show that N₂ has different effects on island nucleation, growth, and coalescence. These effects can be leveraged separately or combined to design versatile manipulation strategies for metal-contact fabrication in which N₂ is introduced to the sputtering...
atmosphere with high temporal precision to selectively target film-growth stages and mechanisms. In this section we explore the viability of this approach by using the following N2 deployment schemes: (i) N2 is introduced at the early film-growth stages for 20 s (which corresponds to a value Θ ≈ 8 ML) and, subsequently, is pumped out from the deposition chamber, while deposition continues and is completed in a pure Ar atmosphere. (ii) Deposition commences in a pure Ar atmosphere, and after 20 s (again Θ ≈ 8 ML) N2 is introduced during the remainder of growth. The effect of the two N2 deployment schemes on Ag-film morphological evolution is shown in Figure 7, which consists of plots of film resistivity ρ as a function of the nominal thickness Θ for the following conditions: (i) deposition in Ar/N2 atmosphere for 20 s followed by growth in Ar (hollow blue triangles; denoted as [Ar/N2(20 s) + Ar]) and (ii) deposition in pure Ar atmosphere for 20 s followed by growth in Ar/N2 (hollow inverse green triangles; denoted as [Ar(20 s) + Ar/N2]). The ρ vs Θ curves for the Ar-grown (hollow black squares) and Ar/N2-grown (hollow read circles) films, previously presented in Figure 1, are reproduced for reference. The continuous film formation thicknesses Θ_cont for the [Ar/N2(20 s) + Ar] and the [Ar(20 s) + Ar/N2] samples are indicated in the figure.

Deployment of N2 after the first 20 s of deposition ([Ar(20 s) + Ar/N2]) results in Θ_cont = 91 ML; i.e., a more pronounced 3D morphology is obtained compared to the Ar-deposited Ag layer. Moreover, the steady-state resistivity value increases to ρss = 1.7 × 10^−5 Ω·cm, which is substantially larger in comparison to the values for both Ar/N2-deposited (ρss ≈ 1.4 × 10^−5 Ω·cm) and Ar-deposited (ρss = 1.1 × 10^−5 Ω·cm) films. Under the [Ar(20 s) + Ar/N2] deposition scheme, N2 is deployed on a pronounced 3D film-growth front which has been formed by extensive and uninterrupted island coalescence during the early growth stages. The effect of N2 is then to interrupt local epitaxial growth and cause grain refinement on the rough postcoalescence film surface. This leads to higher resistivity than that for Ar-deposited samples. The fact that the film deposited under the [Ar(20 s) + Ar/N2] condition exhibits a larger Θ_cont value than the Ar-grown film indicates that interruption of epitaxial growth on a rough surface also impedes the hole-filling process and promotes further 3D growth.

To confirm the morphological trends observed in Figure 7, we perform ex situ XRR measurements on a-C/Ag/SiO2/Si stacks in which the Ag layers were grown at the same conditions are those listed in Figure 7. Analysis of the reflectivity curves (Figure 8; black circles represent experimental data and red solid lines represent calculated curves) of a-C/Ag/SiO2/Si stacks in which Ag films are magnetron sputter-deposited in gas compositions: (a) Ar; (b) Ar/N2; (c) deposition in Ar/N2 for 20 s followed by growth in Ar (denoted as [Ar/N2(20 s) + Ar]); and (d) deposition in Ar atmosphere for 20 s followed by growth in Ar/N2 (denoted as [Ar(20 s) + Ar/N2]). The roughness of the a-C/Ag interface w_a-C/Ag is also provided for each case in the figure.
value \( w_{C/Ag} = 2.3 \) nm is obtained for the film synthesized under the condition \([Ar(20 s) + Ar/N_2]\). The XRR roughness data described above are consistent with the results from the \( \rho \) vs \( \Theta \) curves shown in Figure 7; i.e., early \( N_2 \) deployment promotes 2D growth morphology, while the presence of \( N_2 \) during postcoalescence growth stages yields roughness buildup at the film growth front. Moreover, we find that the mass density of the Ag layers is very close to the bulk value of 10.5 g cm\(^{-3}\) at all deposition conditions.

4. CONCLUSION

The ability to grow noble-metal films with two-dimensional (2D) morphologies on weakly interacting substrates, including 2D materials and oxides, is essential for the fabrication of high-performance enabling devices. The use of less-noble-metal and gaseous surfactants is a known strategy for manipulating the growth of noble metal layers, but the mechanisms by which surfactant atoms affect the complex structure-forming processes are not understood.

In the present work, we combine \textit{in situ} film growth monitoring, \textit{ex situ} characterization, and optical modeling in the framework of the finite-difference time-domain (FDTD) method to study the effect of nitrogen (\( N_2 \)) gas surfactant on growth evolution of nanoscale silver (Ag) islands and films on silicon dioxide (SiO\(_2\)) substrates, with the purpose of exploring the viability of surfactant-based approaches for metal-contact synthesis. We show that \( N_2 \) presence during the early film-growth stages suppresses the rate of island coalescence, which favors 2D morphology and formation of flat films. Furthermore, for later growth stages beyond coalescence, \( N_2 \) causes interruption of crystal growth, which leads to grain refinement and increases the resistivity of continuous layers. FDTD calculations confirm that monitoring of the localized surface plasmon resonance (LSPR) via spectroscopic ellipsometry can be a powerful tool to track and control in real time the early stage morphological evolution of ultrathin metal films and supported nanostructures on weakly interacting substrates.

Using the insights presented above, we design and implement growth manipulation experiments in which \( N_2 \) is deployed selectively during specific growth stages. Early deployment only during the initial growth stages leads to decrease of coalescence rate and roughness development at the film growth front, without compromising the film electrical resistivity. On the contrary, postcoalescence \( N_2 \) deployment leads to pronounced increase in film roughness. This knowledge opens the way for growth strategies in which surfactant species will be deployed with high temporal and spatial precision to target critical film formation stages and manipulate growth morphologies, without altering the film physical properties. Moreover, targeted surfactant deployment can be used to tune the size of supported 3D nanostructures without the need to employ elevated growth temperatures and postdeposition annealing steps.

## ASSOCIATED CONTENT

1. Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsanm.0c00736.

Description of the finite-difference time-domain method that was used to calculate the optical response of silver clusters; images and analysis of early stages of silver film growth obtained from plan-view transmission electron microscopy; determination of localized surface plasmon resonance position from computed total film reflectivity of periodic cluster spatial distributions; X-ray diffractions and selected area electron diffractograms of continuous Ar- and Ar/N\(_2\)-grown silver films; X-ray photoemission spectra of Ar- and Ar/N\(_2\)-grown silver films (PDF)
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