Uncovering the crystal defects within aragonite CaCO₃

Xingyuan San,⁎,⁎, Mingyu Gong,⁎,⁎, Jian Wang,⁎,⁎, Xiuliang Ma,⁎,⁎, Roberto dos Reis,⁎,⁎, Paul J. M. Smeets,⁎,⁎, Vinayak P. Dravid,⁎,⁎, and Xiaobing Hu,⁎,⁎,⁎

Knowledge of deformation mechanisms in aragonite, one of the three crystalline polymorphs of CaCO₃, is essential to understand the overall excellent mechanical performance of nacre. Dislocation slip and deformation twinning were claimed previously as plasticity carriers in aragonite, but crystallographic features of dislocations and twins have been poorly understood. Here, utilizing various transmission electron microscopy techniques, we reveal the atomic structures of twins, partial dislocations, and associated stacking faults. Combining a topological model and density functional theory calculations, we identify complete twin elements, characters of twinning disconnection, and the corresponding twin shear angle (≈8.8°) and rationalize unique partial dislocations as well. Additionally, we reveal an unreported potential energy dissipation mode within aragonite, namely, the formation of nanograins via the pile-up of partial dislocations. Based on the microstructural comparisons of biogenic and abiotic aragonite, we find that the crystallographic features of twins are the same. However, the twin density is much lower in abiotic aragonite due to the vastly different crystallization conditions, which in turn are likely due to the absence of organics, high temperature and pressure differences, the variation in inorganic impurities, or a combination thereof. Our findings enrich the knowledge of intrinsic crystal defects that accommodate plastic deformation in aragonite and provide insights into designing bioengineering materials with better strength and toughness.

Significance

Nacre exhibits outstanding mechanical performance, which results from coordinated deformation mechanisms synergistically working in characteristic microstructures at multiple length scales. A comprehensive understanding of crystal defects within aragonite is critical for discussing the deformation behavior of nacre on microstructure at the nanoscale through atomic scale. By integrating aberration-corrected transmission electron microscopy, crystallographic analysis, and theoretical calculations, we reveal various crystal defects within aragonite at atomic scale and discuss their potential effects on deformation. Our work will serve as cornerstones for modeling analysis and in-depth discussions on nanoscale deformation mechanisms within nacre. Additionally, these atomic-scale insights will benefit theoretical evaluation of the environmental effect on defect formation, enabling defect control in synthetic aragonite and designing of stronger and tougher bioengineering materials.
efficient energy dissipation that delays catastrophic fracture (7). In situ transmission electron microscopy (TEM) nanoindentation testing revealed that the preexisting \{110\} TBs do not migrate (7). Meanwhile, some nanotwins under the indenter are gradually destroyed while other nanograins form due to associated structural transformations in severe local deformation zones. Under high-strain-rate (strain rate \(\sim 10^3/s\)) uniaxial compression, Huang et al. reported two deformation mechanisms—partial dislocation emission and deformation twinning—in aragonite nanoparticles (15). However, they did not reveal the glide plane and Burgers vector associated with partial dislocations and the twin elements associated with the deformation twinning except the twin plane \(K_1 = \{110\}\), although in the latter the same group and their collaborators attempted to explore the twinning shear process using density functional theory (DFT) calculations (16). Thus, to date, a complete description of deformation twinning and partial dislocations is missing, which limits researchers’ deep understanding regarding the intrinsic deformation mechanisms in aragonite.

Here, we uncover the glide plane and Burgers vector of a newly discovered partial dislocation and the complete twin elements of \(f_{110} g\) twinning by integrating aberration-corrected scanning TEM (STEM), crystallographic analysis, and first-principles DFT calculations. In addition, we propose a hitherto unreported potential energy dissipation mechanism in aragonite, namely deformation-induced nanograin formation via pile-ups of partial dislocations. Our findings provide insights into structural defects and deformation behavior in aragonite, which will serve as cornerstones for more robust modeling analysis and explorations on the potential external factors (e.g., environmental pressure, temperature, \(\text{CO}_2\) concentration, and trace elements) which can be used to tailor the defect density. Benefiting from these atomic scale insights, researchers can further explore the exact relationship between the growth conditions and defect density (17–19) and consequently may be able to design stronger and tougher bioengineering materials through the control of defect density within synthetic aragonite.

**Results**

**Stacking Faults and Partial Dislocations in Aragonite.** The macroscopic features and microstructural character of *Pinctada maxima* revealed by X-ray diffraction (XRD) are shown in SI Appendix, Fig. S1. The diffraction peaks of the crushed powders are in good agreement with the aragonite crystal structure with space group \(Pmcn\) and lattice parameters of \(a = 4.96\) Å, \(b = 7.97\) Å, and \(c = 5.74\) Å. The as-obtained *P. maxima* (without crushing the shells) displays evident texture features along the \(002\) plane. This indicates that the broad surface of *P. maxima* shown in SI Appendix, Fig. S1A should approximately correspond to the \(002\) plane. Thus, we prepared electron transparent samples along two distinct directions. One is perpendicular to the broad surface and the other one is parallel to the broad surface. Detailed structural features are revealed in SI Appendix, Fig. S2 and Fig. 1. Along the [100] orientation, which is parallel to the broad surface, we can clearly see the representative brick-wall structures of aragonite as shown in SI Appendix, Fig. S2A. The length of each platelet is around several micrometers, while the thickness is around 200 nm. Serial

![Fig. 1](https://doi.org/10.1073/pnas.2122218119)
electron diffraction patterns shown in SI Appendix, Fig. S2 B–E obtained at different tilt angles can be well-indexed based on the aragonite structure using the above determined lattice parameters using XRD. Along the [001] direction, which is perpendicular to the broad surface of P. maxima, we can see the features of columnar grains as shown in Fig. 1A. Within the higher-magnification images shown in Fig. 1B and SI Appendix, Fig. S3, there are plenty of planar defects on (110) and (110) planes that form an intersection angle of around 64°. These planar defects include both stacking faults and twins (SI Appendix, Fig. S3). An atomic-resolution high-angle annular dark field (HAADF) image and corresponding geometrical phase analysis (GPA) of a stacking fault structure is shown in SI Appendix, Fig. S4. The stacking fault plane can be determined as {110}. There is a very slight, nearly negligible atomic misalignment across the faulted plane. A high-resolution TEM (HRTEM) image in Fig. 1C shows {110} TBs and steps along the TB, respectively, as confirmed by the fast Fourier transform (FFT).

To understand the {110} stacking fault structures, we calculated the γ-surface of the (110) plane using first-principle DFT calculations. The crystal structure of aragonite and associated projections along some low index directions are shown in SI Appendix, Fig. S5. The models shown in Fig. 2 A–C with 24 Ca atoms, 24 C atoms, and 72 O atoms adopt the coordinates with the x axis along the [110] direction, the y axis normal to the (110) plane, and the z axis along the [001] direction. The models have a dimension of 0.947 × 3.553 × 0.581 nm³. Fig. 2A1 shows the single crystal in the view normal to the x axis. The crystal is equally divided into R1 and R2 regions with respect to positions along the z axis. Fig. 2 A2 and A3 show R1 and R2 regions in the view normal to the z axis. The carbonates are marked by triangles, the corners of which denote positions of O atoms. Thus, carbonates are marked by triangles pointing down in the R1 region and by triangles pointing up in the R2 region. In each region, all Ca atoms or carbonates in the same color (black or purple) have the same z position. In a single-crystal configuration, the yellow solid line is parallel to the (130) plane. The green dashed line across black triangles and the blue dashed line across purple triangles are parallel to the (010) plane. The black and gray dashed lines are parallel to (110) shear plane. The black dashed line between two nearby Ca layers is denoted as (110)γ1 while the gray dashed line between two Ca layers separated by one carbonate layer is denoted as (110)γ2.

The γ-surface of (110)γ1 plane in Fig. 2D1 shows two translational vectors, [100] and [110], associated with lattice shear. Fig. 2D2 shows the energy profiles along the [100] and [110] directions on (110)γ1 plane. It is noted that the [110] shear is decomposed into two 1/2[110] shears, i.e., a full dislocation with the Burgers vector [110] which can be dissociated into two partial dislocations with a Burgers vector 1/2[110]. Correspondingly, Fig. 2B shows the fully relaxed stacking fault (SF) structure achieved by 1/2[110] shear on the (110)γ1 plane, referred to as SFCO. The associated SF energy (SFE) is 49.6 mJ/m². It should be noted that the sublattice of Ca atoms does not show faulted stacking. The faulted structure is only associated with the faulting of carbonates. Therefore, the SFCO can only be observed at low magnification in TEM mode due to the local strain contrast along the SFCO. However, when imaging at atomic resolution, the SFCO cannot be observed since faulted C and O atoms with relatively low atomic number do not have sufficient contrast. This is consistent with our experimental observations shown in SI Appendix, Fig. S4.

In addition to the above-mentioned SFCO, there is another type of possible SF. Fig. 2E shows the energy profile along [110] direction on (110)γ2 plane. The two metastable points at 1/2[110] and 0.28[110] correspond to two SFs. Fig. 2C1 shows the faulted structure achieved by 1/2[110] shear on (110)γ2 plane, which is equivalent to the SFCCO as shown in Fig. 2B. There is another possible faulted structure created by 0.28[110] shear on the (110)γ2 plane, referred to as SFCa. After full relaxation, the SFE associated with SFCa is calculated as 426.8 mJ/m². It should be noted that both Ca atoms and carbonates show faulted stackings, implying that SFCa can be observed in atomic-resolution images.

Compared to the SFEs of two possible stacking faulted structures, SFE associated with SFCO is comparable to the SFE of Cu (~41 mJ/m²) which contains a high density of growth twins and SFs. Thus, the high density of SFCa may form during the growth process of nacre and is likely associated with the glide of 1/2[110] partial dislocations during mechanical loading. However, a lower density of SFCa may only form during growth and mechanical loading of nacre because of its high SFE and greater energy barrier.

**{110} Twinning in Aragonite.** We further obtained atomic-resolution HAADF-STEM images of several twin structures along the [001] direction of aragonite. Since the HAADF imaging technique provides atomic number (Z) contrast with intensities proportional to Z 1.7–1.9 (20–22), the bright contrast columns indicate the positions of Ca atoms and light elements such as C and O are invisible in the images. Ca atoms form a pseudo-hexagonal close-packed structure, which poses some difficulty in distinguishing it from a perfect hexagonal crystal in atomic-resolution images. Fig. 3A shows an atomic-resolution HAADF image of a flat TB and Fig. 3B shows a two-layer-thick twin structure. The twinning plane is determined to be {110} according to the selected area diffraction patterns and digital FFT patterns (Fig. 1C and SI Appendix, Fig. S3B). In order to accurately describe the twin structure, five quantities including K1, η1, K3, and η2 and the corresponding twinning angle γ are needed. Based on experimental observations, the quantity K1 can be determined as (110). However, the other quantities cannot be identified by only carrying out the experiment. Fig. 3C shows a twin with one step and the step height is one atomic layer d0. Fig. 3D shows a twin with two steps as clearly revealed in SI Appendix, Fig. S6B by strain analysis using GPA. The step height is approximately seven to eight atomic layers and the magnitude of the net Burgers vector associated with the step is about 1/2[110] according to the Burgers circuits 1 and 2. Additionally, Fig. 3E, an HRTEM image along the [001] direction, shows another twin structure with irrational boundaries. Similarly, dislocations are frequently revealed along the TBs and the net Burgers vector can be determined as 1/2[110] according the Burgers circuits drawn on Fig. 3F. The distance between neighboring dislocations is around 3.8 nm, which is approximately seven to eight atomic layers.

In order to gain a deeper understanding of the {110} twinning in aragonite CaCO₃, we performed DFT calculations and elaborate crystallographic analysis. Fig. 4A is a 2 × 1 × 1 periodic image showing a matrix crystal (lower part), a twin crystal (upper part), and a K1 TB in the center. The twin crystal is rotated 180° of the matrix crystals about the normal of the twin plane. The K1 TB is fully coherent and is thus referred to as a coherent TB (CTB). The model with 20 Ca atoms, 20 C atoms, and 60 O atoms adopts the coordinates of the x axis along the [110] direction, the γ axis normal to the (110) plane, and the z axis along the [001] direction to have a dimension of 0.947 × 2.553 × 0.581 nm³. The dotted black and red rectangles represent the primitive cells in matrix and twin. Each
The primitive cell contains 4 Ca atoms, 4 C atoms, and 12 O atoms. Crystallographic analysis was performed using the topological model (23–25). Fig. 4 B and C show the dichromatic pattern and complex associated with (110) twinning with the coordinate, the same as that of the atomic model (Fig. 4 A). In Fig. 4 C, the black and red symbols represent atoms in matrix and twin, respectively. Circles represent Ca atoms while triangles represent carbonates. Three corners of the triangle correspond the positions of three O atoms of a carbonate (CO₃), and one C atom occupies the center of the triangle. The horizontal brown dashed line represents the CTB. Corresponding to the upward migration of the CTB, atoms/lattices in the matrix orientation will be moved to atoms/lattices in a twin orientation through shear and shuffle associated with the glide of twinning disconnections (TDs). Burgers vectors of possible one-layer TD $b_{\frac{1}{2}[110]}$ and two-layer TD $b_{\frac{2}{3}[110]}$ are defined in the dichromatic pattern (Fig. 4 B), which are 0.44 $[110]$ (with magnitude of 0.42 nm) and 0.12 $[110]$ (with magnitude of 0.11 nm), respectively. TDs with higher step can always decompose into $b_{\frac{1}{2}[110]}$ and $b_{\frac{2}{3}[110]}$. Since the magnitude of $b_{\frac{1}{2}[110]}$ is much larger than $b_{\frac{2}{3}[110]}$, nucleation of $b_{\frac{2}{3}[110]}$ should be easier. DFT calculations further demonstrated that a two-layer-thick twin (Fig. 2 D) is stable while a one-layer-thick twin is metastable.
twin is unstable and will eventually further develop into a two-layer twin structure. So, \(1/2\{110\}\) twinning should be accomplished by nucleation and glide of two-layer TDs \(b_{2/2}\). Therefore, the twin shear direction \(\eta_1\) is \([110]\), the Burgers vector of TDs \(b\) is 0.12\([110]\), and the step height \(h\) is 2. The associated characteristic twin shear is thus calculated to be \(s = |b|/h = 0.14\). The \(K_2\) plane is determined to be \([130]\) planes as indicated by the solid lines (yellow and white) in the matrix and twin domain in Fig. 4B. The angles between two \([130]\) planes on two sides of the CTB is 172.6°, thus the shear angle \((2\alpha)\) is 7.4°, which agrees well with the experimental observation shown in Fig. 3A. The conjugate twin shear direction \(\eta_2\) is \([1310]\). The interface energy of CTB is calculated based on the model shown in Fig. 4A, which is 5.01 mJ/m\(^2\). The extremely low TB energy can well account for the formation of massive growth twins as revealed in Fig. 1B and C.

Crystallographic analysis and DFT calculations confirm that \([110]\) twinning is operated by successive nucleation and glide of two-layer TDs. Thus, the net Burgers vector \((1/2[110])\) associated with the step can be understood as the pile-up of TDs. The ratio of \(b/f_{2/2} = 4.27\), which corresponds to the pile up of at least four TDs. The corresponding height is eight atomic layers, which is consistent with the step height in Fig. 3F. A coherent dichromatic pattern in SI Appendix, Fig. S6C shows stress-free twin steps with one \(1/2\{110\}\) dislocation which was formed by eight \([110]\) layers. In this sense, the mechanism associated with \([110]\) twinning in aragonite CaCO\(_3\) is different from the mechanism associated with successive glide of Shockley partial dislocations on \{111\} twin planes in face-centered cubic metals. Plasticity associated with twinning could be accomplished by thickening growth twins or nucleating deformation twins via nucleation and glide of \(b_{2/2}\) TDs on the TB.

**Discussion**

The possible slip systems of \([110][001]\) were proposed based on the crystal structure and speculation on deformation behaviors. Intuitively, the shortest translation vector in aragonite is \([100]\) with the magnitude of \(\sim 0.5\) nm. The \([001]\) direction has a larger distance with the magnitude of \(\sim 0.57\) nm. The \([110]\) direction was never considered to be a possible shear vector because it is too large, about 0.94 nm. However, the unique partial dislocations on the \([110]\) glide plane with Burgers vector \(1/2\{110\}\) identified here have a shorter magnitude of \(\sim 0.47\) nm. More importantly, the stacking fault structure associated with the glide of the partial dislocation has a lower fault energy \(\sim 49.60\) mJ/m\(^2\). This should account for the presence of SFs. Moreover, it is speculated that some boundaries may form associated with pile-up of partial dislocations in aragonite if the unique partial dislocations are activated during deformation. Interestingly, we observed such kinds of grain boundaries (GBs) associated with formation of nanosized grains (Fig. 5A and SI Appendix, Figs. S7 and S8). The misorientation angle is less than 10° and the misorientation axis is the \([001]\) zone axis.

![Figure 3](https://example.com/fig3.png)

**Fig. 3.** Atomic-scale observations of the twinning structure and associated dislocations on TB. Atomic-resolution HAADF image of (A) the bulk twin structure and (B) a two-layer twin structure along the \([001]\) direction. The rectangle frames in A and B outline the unit cell in aragonite. The twin angles are indicated as well. Atomic-resolution HAADF image of the twin with (C) one layer thickness and (D) varied thickness. The dislocations associated with TB are indicated using the Burgers vector circuits. (Inset) FFT patterns correspond to a twin structure of aragonite. The Burgers vector is determined as \(1/2\{110\}\). (E) HRTEM image of a twin structure with an irrational TB. (f) Local magnification of TB area in E showing the dislocations on the boundary.
Fig. 5A shows a 20-nm nanograin. The atomic-resolution HAADF images shown in Fig. 5B and C reveal structural details of the GB. The digital FFT patterns shown in Fig. 5B, Inset indicates a \( \sim 6.8^\circ \) deflection across the GB highlighted by shaded curves between the (110) planes. Along the GB, an array of dislocations can be identified. Some cores of the dislocations can be well revealed due to a pure rotation operation along the [001] zone axis. As indicated in the Burgers circuits shown in atoms resolved images in Fig. 5C, the Burgers vector of these dislocations can be determined as 1/2[110]. The tilt GB dislocations visualized in atomic-resolution HAADF images (Fig. 5B) are responsible for the in-plane rotation. The average spacing between the adjacent dislocations along the tilt GB is measured to be \( L \sim 3.9 \) nm. According to Frank–Bilby equation, \( \alpha = \sin^{-1}(b/L) \), where \( b \) (~0.47 nm) is the length of the Burgers vector 1/2 <110>. The calculated rotation angle is consistent with our experimental observations. Another representative nanograin structure with a rotation angle of around 9° is shown in SI Appendix, Fig. S7 as well. Thus, we rationalize the formation of nanograins as follows. Corresponding to the characters of GB dislocations in Fig. 5, the nucleation of nanograins at GBs is most likely accommodated by the 1/2[110] partial dislocations. A schematic of the formation mechanisms is shown in Fig. 5D. Arrays of partial dislocations may nucleate at GBs or surfaces and glide on {110} planes toward the grain interior. These partial dislocations pile up and form dislocation walls, which finally result in the formation of a nanograin. Corresponding to this mechanism, a high density of SFs should be present in the nanograin, as observed in the low-magnification HAADF-STEM image (Fig. 5A). However, such SFs cannot be identified in the atomic-resolution HAADF image (Fig. 5B), because the image only shows the sublattice of Ca.

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All above-mentioned crystal defects are obtained from biogenic aragonite materials. To compare with the structural defects of abiotic aragonite, we performed further studies on the geological aragonite minerals which originate from Morocco. The general structural features are shown in SI Appendix, Fig. S9. It is found that a sporadic twinning structure which has the same feature as the one in the biogenic structure indeed existed in the abiotic mineral, but its twinning density appears very low in the abiotic mineral (SI Appendix, Figs. S9 and S10). In biogenic aragonite materials, the representative feature is the existence of the organics. High-magnification HAADF images (SI Appendix, Fig. S11 A and B) demonstrate clear contrast inhomogeneity. The dark area along the boundary and small areas with dark contrast in grain interior indicate a relative lower average atomic number. Thus, these areas likely correspond to inter- and intracrystalline organics, respectively (26). A semiquantitative elements analysis (SI Appendix, Fig. S11C) reveals that the darker-contrast area has a relative higher C/Ca ratio in comparison to the brighter area, which is another supportive evidence for the organics. The atomic-resolution TEM image shown in SI Appendix, Fig. S11D shows that the crystallinity of the dark areas is very poor, which is reasonable considering the existence of intracrystalline organics. In previous work, researchers have demonstrated that the organics play a critical role in defect formation in biogenic aragonite materials. Internal stress, which can result from either the formation of organics within biogenic nacre or volume changes during the transformation process from amorphous CaCO₃ to crystalline aragonite, likely promotes the formation of many defect structures in nacre. Further large-scale theoretical work is still needed to reveal the relationship between organics and defects formation. Besides the biological effect, trace elemental impurities may be another factor that can affect the formation of defect structure within aragonite. Any impurity elements that can reduce the energy of the stacking faults and twin structures will promote the formation of various structural defects in aragonite as well. Therefore, our atomic-scale insights of the defects structure allow researchers to theoretically evaluate the potential effects of the impurity elements on the energy of various defect structures, which may enable researchers to control the defect density within biogenetic aragonite during growth process.

Materials and Methods

Sample Preparation and Electron Microscopy Methods. The biogenic aragonite materials used in the present work are shells of *P. maxima* which were collected from the South China Sea. The abiotic geological aragonite minerals were purchased online and originated from Morocco. To minimize the detrimental effect of collision and dehydration on the structure and mechanical behavior of the shells, they were kept in the refrigerator in the laboratory. To investigate
the characteristics of the defect structures within aragonite mineral, P. maxima were cut along different orientations. Thin foils were prepared with a diamond saw and then mounted on a Mo grid. Then, the foils were mechanically polished using silicon carbide papers to the thickness of ∼30 μm and further dimpled to <10 μm. The electron transparent foils were finally prepared by Ar ion milling in a Gatan precision ion polishing system using a low voltage (3 to 4 kV) to avoid any possible ion beam damage. Then, the electron transparent specimens were coated with carbon (∼30 nm) to enhance the electron conductivity using Carbon Evaporator (Q150RE Plus; Quorum, Inc).

XRD data were acquired using the Bede D1 system with Cu Kα radiation (λ = 0.15406 nm). The selected area electron diffraction patterns, brightfield and dark-field images, and atomic-resolution TEM and HAADF images were obtained using a probe-corrected JEOL ARM 200CF microscope. This microscope equipped with a CEOS probe corrector and the cold-field emission gun was operated at 200 kV. The convergence semiangle used in STEM mode is around 22 mrad and the collection semiangle for HAADF imaging ranged from 90 to 370 mrad. Within this high angle range, the intensity of images is dominated by incoherent Z contrast. Additionally, this microscope was equipped with dual silicon drift detectors with an area of a single detector covering 100 mm². The achievable solid angle for energy dispersive x-ray analysis is around 1.7 sr.

DFT Calculations. For better interpreting the experimental observations, we performed first-principles simulations with the Vienna ab initio simulation package (VASP), with specific emphasis on elucidating the lattice angle differences between the asymmetric TBs.

Optimization of lattice parameters. The generalized gradient approximation with Perdew-Burke-Emzerhof parameterization was used for the exchange and correlation functions (34). The core electrons were replaced by the projector augmented wave pseudopotentials (35) with the valence states 3s3p43d5, 2s2p2, and 2s2p2 for Ca, C, and O, respectively. A 500-eV cutoff on kinetic energy was used for the plane wave basis set. During calculations, the self-consistent iteration was stopped when the change of total energy was smaller than 10⁻⁵ eV. The convergence criterion of geometry optimizations was that the forces acting on each atom is smaller than 0.01 eV/Å. With the first Brillouin zone for unit cells of aragonite CaCO₃ sampled by 9 × 6 × 8 Monkhorst-Pack (MP) K-point grids (36), the optimized lattice parameters are a = 5.016 Å, b = 8.035 Å, and c = 5.812 Å and the cohesive energy is −2.467 eV per atom.

Calculation of γ-surface. The calculation of γ-surface on the (110) plane started from the single crystal structure containing 24 Ca atoms, 24 C atoms, and 72 O atoms as shown in Fig. 2A with x axis along the ⟨110⟩ direction, y axis normal to the ⟨110⟩ plane, and z axis along the [001] direction. The model has a dimension of 0.947 × 3.553 × 0.581 nm². Thus, the area (A) of the ⟨110⟩ plane is 0.947 × 0.581 nm². During relaxation, periodic boundary conditions were applied in x and z directions, atoms on the top and bottom ⟨110⟩ layers were fixed, and the other atoms could relax in all directions. The first Brillouin zone was integrated by 4 × 1 × 6 MP K-points. The total energy associated with the relaxed model with perfect stacking is denoted as E₀. Then, the structure containing faulted stacking was created by shearing the single-crystal model on the ⟨110⟩ plane with certain displacement. During relaxation of faulting structures, atoms on the top and bottom ⟨110⟩ layers were fixed while the other atoms could move in the y direction. The total energy associated with the relaxed model with faulted stacks is denoted as Eᵣ. SFE is thus calculated as

$$\text{SFE} = \frac{E₀ - Eᵣ}{A}$$

A 10 × 6 grid was applied to the ⟨110⟩ shear plane, and SFE associated with each grid point is calculated and shown in Fig. 2D and E.

Calculation of interface energy of CTB. To calculate the interface energy associated with CTB, model 1 containing one CTB and two ⟨110⟩ free surfaces (as shown in Fig. 4A) and model 2 only containing two ⟨110⟩ free surfaces were constructed. Both models contain 20 Ca atoms, 20 C atoms, and 60 O atoms, adopt the coordinates that x axis along the ⟨110⟩ direction, y axis normal to the ⟨110⟩ plane, and z axis along the [001] direction, and have dimensions 0.947 × 2.553 × 0.581 nm². Thus, the area (A) of CTB is 0.947 × 0.581 nm². During relaxation, periodic boundary conditions were applied in x and z directions, atoms on the top and bottom ⟨110⟩ layers were fixed, and the other atoms could fully relax. With the first Brillouin zone integrated by 4 × 1 × 6 MP K-points, the total energies were calculated as E₁ and E₂ for models 1 and 2. The interface energy of CTB (ECₜₜ) is calculated as

$$\text{ECₜₜ} = \frac{E₁ - E₂}{A}$$

Data Availability. All study data are included in the article and/or SI Appendix.

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Author affiliations: *Hebei Key Laboratory of Optic-electronic Information and Materials, The College of Physics Science and Technology, Hebei University, Baoding 071002, China; †Shenyang National Laboratory for Materials Science, Institute of Metal Research, Chinese Academy of Sciences, Shenyang 110016, China; ‡State Key Laboratory of Metal Matrix Composites, School of Materials Science and Engineering, Shangh hai Jiao Tong University, Shanghai 200240, China; §Department of Mechanical and Materials Engineering, University of Nebraska-Lincoln, Lincoln, NE 68583; †Department of Materials Science and Engineering, Northwestern University, Evanston, IL 60208; and ¶The Northwestern University Atomic and Nanoscale Characterization Experimental Center, Northwestern University, Evanston, IL 60208.
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