Facile design and development of nano-clustery graphene-based macromolecular protein hydrogel loaded with ciprofloxacin to antibacterial improvement for the treatment of burn wound injury

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Abstract
Nowadays, awareness about the burn wound is often considered difficult due to bacterial and other organism infections. The facile and eco-friendly preparations of antibiotic-loaded hydrogel-based bio-composites have great attention in the field of wound dressing for burn wound therapy and nursing care. In the present investigation, we have developed ciprofloxacin (CF)-encapsulated graphene-silk fibroin macromolecular hydrogel dressings material with unique chemical and physical properties to achieve the desirable antibacterial efficacy and healing activity. The antibacterial activity of prepared hydrogel was evaluated against bacterial pathogens treated with different concentrations of CF, which have been provided improved antibacterial activity on burn wound infection. In vitro, cytocompatibility evaluations were performed to imply the suitability of hydrogel on fibroblast cells, which has been dramatically related to in vivo wound healing. Furthermore, an in vivo wound healing analysis was carried out using a rat to observe the capability of the CF-incorporated GH/SF hydrogel matrix. Thus, this investigation widely demonstrates the healing ability of prepared hydrogel matrix and could be a significant landmark in the research on burn wound healing applications.
Keywords  Burn wound · Graphene nano · Silk fibroin · Hydrogel · Ciprofloxacin · Antibacterial activity

Introduction

Burns is one of the worst forms of the outer layer of skin trauma and remains a serious global health issue. Burn wound is quickly susceptible to bacterial infection in skin tissue; it may delay the epidermal maturation and leads to surplus scar formation, organ failure, and death [1, 2]. Inflammation, cell proliferation, and subsequent tissue formation are complex processes that the bacterial invasion mainly caused by *Pseudomonas aeruginosa* may delay the repairing and reconstructing burnt wounds [3]. To avoid these complications, we have to develop multifunctional antibiotics to protect the wound from an invasion of microorganisms to reduce infection, promote skin regeneration and tissue re-formation [4].

Several nontraditional antibacterial agents, such as metal, metal oxide NPs, carbon nanomaterials, and graphene-based materials, have been exploring in biomedical science. Among these materials, graphene-based materials have received plenty of attention for numerous applications, including biological that can provide a suitable scaffold for the wound healing process, which can improve the granulation tissue thickness, density of fibroblasts, collagen, and vascularization because of their stimulating effect on cell proliferation and electroconductivity properties [5–8]. Graphene is a building block for several carbon materials with no defect, and it consists of a single layer 2D structure. Therefore, graphene emerged as a promising material with its unique properties such as excellent mechanical strength, biocompatibility, hydrophilic groups, and surface area [9, 10]. The antibacterial and antifungal properties of graphene inhibit the wound from infection can potentially improve the wound healing process. Most promisingly, the graphene is used against corona virus
porcine epidemic diarrhea when used a minor component of composite material [11]. Meanwhile, GO loading with active antibacterial substances shows superior antibacterial and dispersing effects [12]. On the other hand, physical damages of bacteria have been caused by graphene through direct contact of its sharp edges, and the chemical mechanism of graphene-attached cell membrane could induce oxidative stress [9, 13].

Natural polymers such as protein, peptides, carbohydrates, and polysaccharides have excellent biocompatibility that has been utilized for wound dressing. In particular, silk fibroin (SF) is a natural polymer that can activate the NF-κB signaling, closely relating to the cutaneous wound healing process. The excellent biocompatibility, improved mechanical property, non-cytotoxicity, low antigenicity, and non-inflammatory properties of SF facilitate the enhanced cell adhesion, proliferation, tissue regeneration, and matrix deposition of keratinocytes and fibroblasts [14, 15]. Besides, it is very promising to control bacterial infection in different research studies and accelerated wound healing properties [16]. In addition, the degradation product of silk protein matched with body metabolism [17]. Various forms of silk fibroin have been used for wound dressing applications, including films, fiber, hydrogel, and sponges [18]. According to the characteristic structure of SF-based hydrogel with some unique properties such as adhesiveness and anti-oxidant to facilitate self-healing [7, 19], the connection with antibiotics is an initial stage of research for improving wound healing [20]. Antibiotics are generally used to prevent bacterial infection and biofilm formation on skin wounds. To date, many antibiotics have been reported to inhibit bacterial infections such as clindamycin, gentamicin, minocycline, and so forth [21–23]. Among the antibiotics, ciprofloxacin belongs to the fluoroquinolone family. It is one of the extensively used antibiotics for treating wound healing against in vitro bacterial infection; it’s superior to other antibiotics in clinical practice, while their spontaneous bacterial resistance is a meager and lesser incidence of bacterial resistance [24, 25]. Therefore, we need to fabricate the ciprofloxacin-loaded biodegradable hydrogel composite for wound dressing applications [26, 27].

Herein, we report the fabrication of CF-loaded nano-graphene/SF hydrogel-based biocomposite material an effective multifunctional antibacterial agent for burn wound treatment. The structural, morphological, and thermal properties were analyzed. The antibacterial efficiency of hydrogel-based biocomposite was evaluated. Finally, that synthesized hydrogel composites are effective for wound healing with enhanced antibacterial activity (CF) for preventing bacterial infection.

**Materials and methods**

**Materials**

Graphite flakes, sulfuric acid, hydrochloric acid, hydrogen peroxide, potassium permanganate, sodium carbonate dehydrous (Na₂CO₃), lithium bromide 99%, ciprofloxacin (CF), and ascorbic acid were purchased from Sinopharm Chemical Reagent Co., Ltd (Shanghai, China). Deionized water was used throughout the experiment.
Preparation of silk fibroin solution

Silk fibroin was extracted according to the previous literature from Bombyx mori silk cocoons. First, the cocoons were boiled with aqueous Na₂CO₃ at 85 °C for removing sericin. After the obtained degummed silk, fibroin was dissolved in 9.3 M LiBr solution at 60 °C for 2 h. Then, the solution was dialyzed (MWCO 8000) against distilled water and centrifuged. The supernatant was stored at 4 °C. Finally, the concentration of SF solution was measured at the weighing method [28].

Preparation of graphene/silk fibroin

Modified Hummer’s method was applied to prepare graphene oxide (GO) from graphite flakes [29]. The dispersed graphene oxide was vigorously stirring for 30 mts at 110 °C and sonicated for 1 h to get aqueous dispersion of GO. After that, ascorbic acid was added into the above mixture to get a proper reduction at 80 °C for 24 h. Next, 100 mg of silk fibroin (SF) was dissolving in 10 ml phosphate buffer (pH 4) under constant stirring at 60 °C for 4 h. Then, the prepared SF solution was then added into the above mixture with vigorous stirring for 30 mts at 110 °C. The mixture was vortex until to get a homogeneous solution. Finally, the hydrogel was prepared by the ultrasonic method that the hydrogel formed by the self-assembly method. The final mixture was transferred into a beaker and degassed by ultrasound for 30 mts and was regenerated using deionized water. The obtained black-colored hydrogel was thoroughly washed with water and dried.

Preparation of CF-loaded nano-graphene/silk fibroin hydrogel

100 ml of CF solution was added with graphene/silk fibroin solution placed in a water bath shaker for 3 h under dark conditions, and the supernatant solution was trained off. Finally, the CF-loaded graphene/silk protein composites were dried using a hot air oven and were characterized for further studies.

Antibacterial activity

The prepared hydrogel composites were tested by well diffusion method against pathogenic bacteria S. aureus—ATCC 11,632, and P. aeruginosa—ATCC 15,442. The inhibited zone of diameter around the disks was recorded of the test organism. The plates were autoclaved at 121 °C for 15 min. Then, the agar medium was sterilized and transferred to a sterile petri dish. After getting solidified agar, the plates were stored at the refrigerator (4 °C). After freezing the media, the bacteria were diffused using a sterile cotton swab. The petri dishes were incubated for 24 h at 37 °C. After one day, the zone of inhibition of bacteria was measured.
Growth curve

The antibacterial activity of composites was analyzed from bacterial growth inhibition by the standard microdilution method. The bacterial strain was grown in 10 mL of LB media-containing flask. Then, 0–40 µg/mL composites were supplemented with media, and the bacterial culture was incubated overnight under shaking environment at 160 rpm in 37 °C. The bacterial growth was determined by using UV–Vis spectrophotometer (OD 595 nm) with a time interval. The growth curve was plotted against optical density versus time.

Biofilm assay

In biofilm inhibition assay, the biofilm formation was evaluated by CLSM (Carl Zeiss, LSM 710, Germany, software; Zen 2011) using 24-well plates, and bacterial viability was analyzed using Live/Dead Baclight bacteria viability kit. In brief, 100 µL of bacterial cultures (1×10^8 cells/mL) were incubated for 24 h in sterile LB broth with hydrogel. The planktonic cells and exhausted media were discarded after 24-h incubation. The non-adhering bacterial cells were removed using PBS, and each glass slide was stained with 0.01% acridine orange. The stained cell was detected using 488 nm wavelength for excitation and 540 nm for emission, and the images were observed under CLSM.

MTT assay

MTT (3-[4, 5-dimethyl-2-thiazolyl]-2, 5-diphenyl-2H-tetrazolium bromide) assay is used to determine the cell viability and cytotoxicity of the hydrogel. The cells were grown in 24-well plates at 37 °C in a humidified atmosphere of 3–5% CO₂ for 24 h. After 24 h of incubation, the culture medium was discarded from each well, and then prepared hydrogels were transported into new 24-well culture plates (cell density = 2×10^4 cells per well). Then, the assays were incubated with MTT solution at 37 °C for 4 h with a tetrazolium salt solution. The MTT solution was discarded carefully, and 150 µL of 10% dimethylsulfoxide (DMSO) was added into each well to solubilize the formazan crystal. The cell viability of hydrogel materials was evaluated at day 1, 3, and 5 after cell seeding. Each sample was analyzed under three replicates experiment for the exact quantitative value. Finally, the cell viability (%) rate was measured by using the subsequent equation.

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\text{Percentage of cell viability} = \frac{[A]_{\text{treated}}}{[A]_{\text{control}}} \times 100
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In vivo animal experiment

In vivo animal experiments were approved and carried out according to the Department of Science and Education, Jinan People’s Hospital, China. During this animal experiment, 5 weeks of Kunming mice were utilized. At first, each mice were anesthetized using controlled dosage and then 7 mm of partial thickness burns wounds
were formed on back side of mice. After that, the hydrogel was applied in each model. The dressing samples were reformed on the wound every day. The healing rate was measured, and the photograph was taken at 3, 6, 12, and 18 d. For histological examination during the wound regeneration, each model was stained with hematoxin–eosin (H&E) and photographed using a microscope.

Characterization

Thermogravimetric analysis of hydrogel composites was conducted on a Netzsch TG instrument under nitrogen atmosphere and the heating rate of 10 °C min⁻¹ in the temp range from room temp to 700 °C. XPS was performed using an ESCAL-AB250Xi photoelectron spectrometer (ThermoFisher Scientific, USA) with X-ray radiation. The crystalline structure of hydrogel was examined by X-ray diffractometer (PAN analytical X'pert PRO) at 30 kV, and 40 mA was used as X-ray source. Raman spectra (LabRAM HR800, Horiba Jobin Yvon) were recorded to investigate the crystalline structure. SEM analysis was performed Carl Zeiss MA15/EVO 18). FTIR spectra of hydrogel materials were recorded by Alfa Bruker FTIR Spectrophotometer.

Result and discussion

X-ray diffraction patterns of SF, graphene, SF/graphene, and hydrogel composites are shown in Fig. 1. The SF shows two peaks at 20° and 24°, which are related to the β-sheet structure [30, 31]. In graphene, the broad peaks appearing at around 2θ = 25.5° and 43.3° are corresponding to (002) and (100) planes (Fig. 1b). The pattern was well matched with JCPDS NO: 75-1621 with hexagonal structure with sp2-bonded carbon atoms [32]. Meanwhile, the XRD patterns of the SF/
graphene can be exhibited the characteristic peak of SF, which indicates that the addition GO does not affect the crystalline structure of SF due to the strong interaction between the SF and graphene [33]. Figure 1d shows the CF-loaded SF/graphene displays a number of characteristic crystalline peaks is observed.

The morphology of SF, graphene, SF/graphene, and hydrogel composites was illustrated by SEM images and is presented in Fig. 2. The pure SF has depicted the interconnected macroporous structure, and a pore is distributed [31]. Graphene nanosheet consists of randomly aggregated, and cross-linked porous structure is shown in SEM images [29, 34]. The synthesized graphene/SF shows that the graphene sheet was robust and no exfoliation surface indicates that the nanosheet was well dispersed into a polymer matrix via polar–polar, H-bonding, and hydrophobic–hydrophobic interaction. As shown in Fig. 2b, a wrinkled film was covered on the fiber surface; at the same time, SF obtained a smooth surface without pleats [35].

The FTIR spectrum of SF, graphene, SF/graphene, and hydrogel composites is shown in Fig. 3a. In Fig. 3a, the characteristic peak at 3253 cm$^{-1}$, 1625 cm$^{-1}$, 1522 cm$^{-1}$, and 1234 cm$^{-1}$ represents the OH stretching, amide I, amide II, and amide III of β-sheets and the random coil of SF [36, 37]. In Figure 3b, the characteristic bands of surface oxidized groups at 3450 cm$^{-1}$, 1384 cm$^{-1}$, and 1028 cm$^{-1}$ indicated the stretching vibration of O–H, C–O–H, and C–O, respectively. The peak at 1635 cm$^{-1}$ could be attributed to the skeletal vibration of graphitic carbon [34, 38]. For the SF/graphene, the spectra (Fig. 3c) presented the peaks of graphene at 1371 and 1589 cm$^{-1}$ and the peak of SF at 1523 cm$^{-1}$ and 1647 cm$^{-1}$ strong, which revealed the successful hybridization of SF/graphene matrix [39]. The characteristic band at 1588 cm$^{-1}$ and 1540 cm$^{-1}$ from the benzene ring stretching vibration and the band at 3200 cm$^{-1}$ and 1612 cm$^{-1}$ are related to O–H and C–O stretching of CF-loaded SF/graphene [40]. Almost all the main peaks of the ciprofloxacin are detected into prepared hydrogel, indicating the successful formation of CF in hydrogel, and the peaks are shifted due to the intermolecular hydrogen bonding [41, 42].

![Fig. 2](image-url) The surface morphological structure of the prepared silk fibroin hydrogel in the absence (a) and presence (b) of nano-graphene sheet was exhibited through scanning electron microscopic (SEM) analysis.
Raman spectra of graphene, SF, and SF/graphene hydrogel are shown in Fig. 3b. The Raman shift of SF was consistent with the spectral peaks around 1003 cm\(^{-1}\), 1326 cm\(^{-1}\), and 1658 cm\(^{-1}\), known as the characteristic β-structure [43]. The Raman spectra have prominent peaks observed at 1600 cm\(^{-1}\), attributed to the G band of carbon. The D band is an indication of structural defects or disorder of graphitic domains [29]. The ID/IG ratio is decreased, which denotes that the materials have a degree of disorder due to the introduction of SF in hydrogel [7, 44, 45]. As shown in Fig. 3b, SF/graphene hydrogel displays that the three dominant peaks at 1080 cm\(^{-1}\), 1225 cm\(^{-1}\), and 2943 cm\(^{-1}\) are attributed to the presence of SF [35]. Both the characteristic peaks of graphene and SF were visible from the distinct spectral line of the composites, which demonstrated that presence of graphene/SF.

To evaluate the thermal stability of graphene/SF composites as shown in Fig. 4, the first weight loss occurs at room temperature to 150 °C, which is due to the removal of surface water that indicates the large amount of water bound in SF. After the addition of graphene, water is decreased gradually and reduced to 3.22 wt% for the graphene. All the samples showed that the stability is up to 160–210 °C while above the 210 °C the degradation is started. The rapid weight loss is observed in the pure silk sample, and the graphene has slight weight loss when the temp at 200 °C has higher thermal stability than other samples. Graphene is used in various materials with controlling thermal stability due to its excellent thermal properties [46]. The thermal stability of hydrogel increased when we used graphene into the hydrogel due to the hydrophobic interaction between graphene and SF [45].

The chemical composition of SF/graphene composites was carried out by XPS analysis. Figure 5 shows the characteristic peaks with binding energy 284.6 eV and 533.3 eV, which is attributed to C(1 s) and O(1 s) [47]. In the spectrum of SF, the C 1 s, N 1 s, and O 1 s peaks are centered in the binding energy of 287.1 eV, 399.7 eV, and 532.7 eV [48]. The peak is observed at 684.8 eV in CF-loaded graphene/SF is evidence for F element existence in composites [49].

Fig. 3 The chemical structure and intermolecular interactions between the distributed nano-formulations and bio-polymeric groups have been observed and demonstrated by the FTIR and Raman spectroscopic analysis; a FTIR spectroscopic investigations of prepared SF, GH, GO and CF-loaded GH/SF samples and b Raman spectroscopic analysis of prepared SF, GH, and SF/GH, respectively
The antibacterial activity of CF-loaded graphene/SF hydrogel was examined against *P. aeruginosa* and *S. aureus* is presented in Fig. 6. The zone of inhibition method was used to evaluating the antibacterial properties as a simple and effective method. The wound healing can be invasively affected by *P. aeruginosa*, and *S. aureus* is the most common bacteria for creating a chronic wound. On the other hand, the great challenge of wound recovery directly depends on the intrinsic

**Fig. 4** The thermal stability and nature of temperature on prepared materials were observed by thermal gravimetric analysis (TGA); TGA results exhibited the thermal degradation behaviors of prepared SF, GH, and GH/SF in the temperature from 0 to 700 °C

**Fig. 5** The elemental presentations and nature of prepared hydrogel materials were observed by X-ray photoelectron spectrum (XPS) survey analysis; the presentation of carbon (C), nitrogen (N), and oxygen (O) elements into the prepared SF, GH, and GH/SF hydrogel samples was observed

The antibacterial activity of CF-loaded graphene/SF hydrogel was examined against *P. aeruginosa* and *S. aureus* is presented in Fig. 6. The zone of inhibition method was used to evaluating the antibacterial properties as a simple and effective method. The wound healing can be invasively affected by *P. aeruginosa*, and *S. aureus* is the most common bacteria for creating a chronic wound. On the other hand, the great challenge of wound recovery directly depends on the intrinsic
antibiotic resistance of bacteria. Therefore, some powerful antibacterial agents can have used in biomaterials, particularly in wound dressing. The CF-loaded graphene/SF exhibited distinctive antibacterial effects against \textit{P. aeruginosa} and \textit{s. aureus} with a clear zone of diameter as one of the important infectious microorganisms in wounds. However, low inhibition zone was formed around hydrogel without CF. These results showed the physical disruption of ROS species and sharp edges of GO which lead to remarkable damages [50]. These results suggested that the hydrogel has less zone of inhibition of \textit{P. aeruginosa} and \textit{S. aureus}, while CF-loaded hydrogel can effectively inhibit the growth and replication of \textit{P. aeruginosa} and \textit{S. aureus} because CF implies good release and diffuse out into the surface. These synthesized hydrogel composites have been used as an antibacterial drug delivery system with the controlled release [49].

Figure 6 shows the optical densities of composites and CF-loaded graphene/SF with various dose concentrations (0–40 µg mL\(^{-1}\)) of CF were plotted at regular time intervals. The composites showed concentration-dependent relative rate and extent of bactericidal activity. The hydrogel composites exhibit both bacteriostatic and bactericidal activity because of their better efficiency, and there is a sudden decrease of growth of the bacteria when the concentration is increased therefore avoiding the exogenous infection effectively. Moreover, the antibacterial activity of composites is increased when the amount of CF-loaded graphene/SF is increased, owing to the

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**Fig. 6** Antibacterial properties of prepared CF-loaded GH/SF hydrogel materials were investigated against clinically approved gram-positive (\textit{S. aureus}) and gram-negative (\textit{E. coli}) bacterial pathogens; \textbf{a} & \textbf{b} growth curve analysis of as-prepared CF@GH/SF hydrogel material in different concentrations and zone of inhibition; \textbf{c} & \textbf{d} results observation of prepared CF@GH/SF sample in different concentration
sustained release of CF from composites and sustained antibacterial activity, and the hydrogel could play a reservoir. Therefore, we give proper protection against wounds when the treatment with burn wound dressing [51]. The present study reports that CF-loaded graphene/SF inhibits the bacterial growth in burn wounds from being exogenous infected for several days. Its acts as an effective source of antibacterial biocides.

The cytocompatibility of CF-loaded graphene/SF was investigated against human fibroblast cell line as exhibited in Fig. 7. Generally, the fibroblast cell line is a widely tested cell line for wound healing materials. It is commonly presented human cell lines and incredibly wound healing contributed cell lines in the skincare and wound management processes. The cell migration ability of the hydrogel has been exhibited at different time intervals, as shown in Fig. 7a. In addition, the signaling from the growth factors of fibroblast cells is the main factor of wound ECM synthesis and intracellular mechanisms of skin wound healing biological processes. In the present investigation, the fibroblast cell is treated with hydrogel nanocomposites, enhancing cell survival and proliferation at different time intervals. Figure 7b shows the enhanced cell survival and proliferation (Fig. 7c) treated with hydrogel composites, which imply that the material is significantly influencing the inhibition of bacterial infection and improving cell viability. The above results showed that the enhanced proliferation is the combination of graphene; it may increase oxygen-containing groups [33].

In the normal wound healing process, a traditional dressing method could be suitable and adheres to the injured tissues and creates solutions to form newly produced

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**Fig. 7** In vitro cytocompatibility of prepared hydrogel materials was studied by the methods of quantitative observation of in vitro cell viability (MTT assay) and cell migration analysis in different time intervals; a microscopic visualization of in vitro cell migrations treated with control (PBS), GH/SF, and CF@GH/SF samples in different time durations (6, 12, and 18 h); b & c qualitative results of cell viability and cell proliferation in different time intervals
granulated tissues for the healing. But, the burn wound and other problematic wound healing processes need special attention and tailored materials compositions to provide superior properties such as antibacterial, anti-fouling, anti-oxidant etc. Furthermore, the anti-infectious ability of prepared CF-loaded GF/SF hydrogel matrix has been confirmed by the antibacterial experiments. In addition, in vivo healing ability of the prepared GF/SF and CF@GF/SF hydrogels was investigated on the backside of rats under post-treatment on different days (3, 6, and 12 days) as exhibited in Fig. 8. The healing ability on treated wounds was recorded about the total area covered (healed) in different post-treatment days. On post-treatment day 3, the wound environments were consistent and haven’t observed many changes in all the groups. On post-treatment day 6, it was exhibited that the wound healing percentage of CF@GF/SF and GF/SF hydrogels treated wound site have been higher and extensively covered the injured area when compared to the control samples. On post-treatment day 12, the wound site of GF/SF and CF@GF/SF hydrogel samples was significantly healed and covered the injured area compared to the control sample. These results showed that the antibacterial hydrogel could provide a suitable scaffold for cell attachment, proliferation, and wounds inhibition from infection can potentially improve the wound healing process [7]. The superior healing properties of hydrogel might be attributed to graphene and the sustained release of CF. The CF has excellent antibacterial and anti-inflammatory activity, which could effectively inhibit the bacteria infection on the wound surface, thereby accelerating wound healing [52]. The H&E stained analysis has further evaluated the post-treated wound tissues on
different days through microscopic analysis to observe its host immune responses, as shown in Fig. 9. The control sample shows more inflammatory cells on post-treatment days 6 and 12 when compared to the hydrogel samples, which was confirmed that prepared hydrogel samples provided significant anti-inflammatory response and host immune responses in the healing process. As shown in Fig. 9, after treatment for 12 days, CF@GF/SF hydrogel-treated wound site shows a clear dermis layer, which confirming that successful re-epithelialization when compared to the control and GF/SF hydrogel sample. These results established that CF@GF/SF hydrogel could promote the cell growth of fibroblasts, keratinocytes, and endothelial cells and showed a positive effect for the wound healing process during in vivo wound healing treatment [53]. The superior healing capacity of hydrogel might be attributed to the presence of GO, which promoted cellular proliferation and migration involved in the proliferative phase [52].

**Conclusion**

In summary, we report a synthesized graphene/silk fibroin hydrogel wound dressing material which was further loaded with ciprofloxacin and was characterized. The XRD study explored the synthesis of graphene using graphite powder. The functional group and structural properties are confirmed by FTIR and Raman spectra. The SEM images showed that the nanosheet was well dispersed into the polymer matrix. These composites hydrogel showed enhanced antibacterial activity toward

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**Fig. 9** Histomorphological observation (hematoxylin–eosin staining microscopic study) of burn wound regeneration after treatment of prepared hydrogel dressings of GH/SF, CG@GH/SF, and control sample on the 3rd, 6th, 12th, and 18th day after postoperative; the images exhibited in different magnifications (10x and 40x)
both bacteria. In vitro cytocompatibility studies confirmed that the hydrogel was nontoxic toward the fibroblast cell line. In addition, the in vivo wound healing analysis of hydrogel effectively hastened the wound healing process after burning wound infections. In vitro and in vivo studies have been demonstrated the probable effect on fibroblast adhesion, fastening re-epithelialization, and keratinocyte proliferation. Overall, the CF-loaded graphene/silk protein hydrogel gave significant potential application; it is essential to wound dressing material in the burn healing process.

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