Chemistry of two new Leprariaoid lichens from Sri Lanka

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Abstract: The hexane and dichloromethane extracts of the lichen Leproloma sipmanianum, collected from Beragala, contained the compounds, atranorin 1, β-sitosterol 2, (+)-usnic acid 3, zeorin 4 and tritetracontylpentanoate 5 while the methanol extract furnished the triterpenoid 3β-acetoxyfern-9(11)-ene 6. In addition to compounds 1-5, the lichen powder yielded glyceryl trilinolate 7 and 3,6-dimethyl-2-hydroxy-4-methoxybenzoic acid 8. Chromatography of the lichen powder of Lepraria atrotomentosa led to the isolation of five compounds, including compounds 1-4 along with methyl-β-orcinolcarboxylate 9. (+)-Usnic acid exhibited potent antitermite activity against a common pest of tea at low elevations, Glyptotermes dikztatus. 3,6-Dimethyl-2-hydroxy-4-methoxybenzoic acid 8 showed moderate larvicidal activity against the second instar larvae of Aedes aegypti.

Key words: Anti termite assay, atranorin, 3,6-dimethyl-2-hydroxy-4-methoxybenzoic acid, glyceryl trilinolate, Lepraria atrotomentosa, Leproloma sipmanianum, lichen, methyl-β-orcinolcarboxylate, mosquito larvicidal assay, tritetracontylpentanoate, (+)-usnic acid, zeorin, 3β-acetoxyfern-9(11)-ene

INTRODUCTION

Lichens are biologically distinct entities composed of an algal or cyanobacterial partner (photobiont) and a fungal partner (mycobiont) living in a symbiotic state.¹ The mycobiont plays an important role in absorbing mineral nutrients and protects the lichen from exposure to intense sunlight and desiccation. The mycobiont, in turn, obtains sugars and in some cases organic nitrogen from the photosynthetic partner.² Lichens synthesize a wide variety of metabolites with different structures and potential biological activities. These activities range from being poisonous (to insects, snails and nematodes) to antibiotic, antitumour, antitherbivore and growth and enzyme inhibitory.³ Lichens play a prominent role in air pollution studies because of their sensitivity to different gaseous pollutants, trace metals and radioactive elements.³

METHODS AND MATERIALS

General methods: Melting points were determined by using a Kofler hot stage apparatus and are uncorrected. UV absorptions were measured in a Shimadzu 1601 UV spectrophotometer. IR spectra were recorded on a Shimadzu 160 spectrophotometer on KBr pellets. Optical rotations were determined in a Bellingham-Stanley ADP 220 polarimeter. ¹H NMR, ¹³C NMR/ DEPT, HETCOR and HMBC spectra were recorded on a VARIAN (¹H 300 and ¹³C 75.45 MHz) in CdCl₃ with TMS (tetramethylsilane) as the internal standard. Low and high resolution electron impact mass spectra were recorded on a Kratos/AEI MS-902 spectrometer detector. FAB (Fast Atom Bombardment) mass spectra were measured on a Varian Mat CH 4-B spectrometer using 1-thioglycerol as the matrix. Silica gel used was Merck Kieselgel (230 - 400 mesh ASTM).

Collection, extraction and isolation: L. sipmanianum was collected from Beragala (Uva Province). Specimens were cleaned, air-dried and ground to a fine powder (100 g), which then was sequentially extracted into hexane, dichloromethane and methanol using a bottle shaker at room temperature. The methanol extract (10 g) was subjected to Medium Pressure Liquid Chromatography (MPLC) (eluent: CH₂Cl₂ to MeOH) to yield 12 fractions. The first two fractions upon further MPLC (eluent: CH₂Cl₂: hexane (1: 4) CH₂Cl₂) gave 3β-acetoxyfern-9(11)-ene 6 (68.7 mg) which was crystallized from CH₂Cl₂ while fractions 3 and 4 of the CH₂Cl₂ to MeOH MPLC yielded atranorin 1 (Figure 1).
The lichen powder upon MPLC {eluent: CH₂Cl₂: hexane (1: 39) to MeOH} yielded 6 fractions. Fraction 2 upon gravity column chromatography {eluent: CH₂Cl₂: hexane (1: 17)} gave triteracontylpentanoate 5 (1 mg). Fraction 3 of the same column was subjected to further MPLC {eluent: CH₂Cl₂: hexane (1: 9) to CH₂Cl₂: hexane (3: 2)} and furnished 3, 6-dimethyl-2-hydroxy-4-methoxybenzoic acid 8 (19 mg); fraction 6 upon MPLC {eluent: CH₂Cl₂ to MeOH: CH₂Cl₂ (1: 19)} yielded zeorin 4 (89 mg).

L. atrotomentosa was collected from Ramboda (Central Province). Specimens collected were cleaned, air-dried and ground into a fine powder. It was mixed with silica gel and subjected to direct MPLC {eluent: CH₂Cl₂: hexane (1: 9) to CH₂Cl₂: hexane (3: 2)} to yield 5 fractions. Fraction 3 upon MPLC {eluent: CH₂Cl₂: hexane (2: 3) to CHCl₃} gave atranorin 1 (63.7 mg) and β-sitosterol2 (1.6 mg); fraction 5 upon repeated MPLC yielded methyl-β-borcinokarboxylate (CHCl₃) from fraction 4. Fraction 5 upon repeated MPLC yielded methyl-β-borcinokarboxylate 9 (1.3 mg).

**Hydrolysis of triteracontylpentanoate 5:** Triteracontylpentanoate 5 (10 mg) was dissolved in THF (tetrahydrofuran), mixed with 10% NaOH (3 ml) and stirred for two h. The reaction mixture was monitored by TLC (Thin Layer Chromatography) for the disappearance of the starting material. After the completion of the reaction, the mixture was acidic. The solid obtained was filtered and the filtrate was extracted with CH₂Cl₂ (10 ml x 3). The CH₂Cl₂ was evaporated in a rotovapor and the resultant crude solid was used to obtain the 'H NMR spectrum.

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**Atranorin 1:** Colourless needles (CH₂Cl₂); m.p. 195-196°C (lit. m.p. 196°C).β

**β-Sitosterol 2:** Colourless needles (CH₂Cl₂); m.p. 140-142°C (lit. m.p. 140-141°C).6

**(+)-Usnic acid 3:** Yellow prisms (CH₂Cl₂); m.p. 203-204°C (lit. m.p. 203°C).

**Zeorin 4:** Colourless needles (CH₂Cl₂); m.p. 234-236°C (lit. m.p. 234-236°C).7

**Methyl-borcinokarboxylate 9:** Colourless needles (CH₂Cl₂); m.p. 141-142°C (lit. m.p. 140-141°C).8

**Fern-9(11)-ene 3:** Colourless needles; m.p. (274-276)°C; [α]D = 257° + 83.3° (CH₂Cl₂); 1H NMR (300 MHz, CDCl₃); δ 1.28, 1.38 (6H, overlapping t, J = 5.1 Hz, Me), 1.25 (82H, bs, CH₂), 1.60 (4H, m, CH₂), 2.30 (2H, t, J = 7.5 Hz, CH₂), 4.05 (2H, t, J = 10.5 Hz, CH₂); EIMS m/z 704, 676, 648, 620, 592, 564, 536, 465, 420, 392, 369, 341, 313, 295, 285, 257, 111, 97, 85, 71, 57 (100), 34; HREIMS (M⁺) m/z 704.7365 (calcd. for C₃₈H₅₀O₆, 704.7403).

**3, 6-Dimethyl-2-hydroxy-4-methoxybenzoic acid 8:** Pale yellow needles; m.p. 195-196°C (lit. m.p. 196°C).9

**Methyl-borcinokarboxylate 9:** Colourless needles (CH₂Cl₂); m.p. 141-142°C (lit. m.p. 140-141°C).10
Mosquito larvicidal assay\textsuperscript{13}: A laboratory culture of *Aedes aegypti* was dissolved for the assay. Test solutions were prepared by dissolving crude extracts (100 mg) or pure compounds (1 mg) in acetone (1 ml) and diluting with water up to 200 ml (500 ppm for extracts; 10 ppm for pure compounds). Polyethylene glycol (120 µ1) was added to homogenize the test solutions. These test solutions were used as treatments to screen against the second instar larvae of *A. aegypti*. Four replicates (five larvae in each beaker) were used for each treatment and the control contained all the ingredients except the lichen extract or the pure compound. The number of larvae moribund or dead was recorded after 24 and 48 h.

Antitermite assay\textsuperscript{14}: Test compound (10 mg) was dissolved in a minimum amount of CH\textsubscript{2}Cl\textsubscript{2} and mixed with cellulose (1.25 g). The mixture was air dried under a current of warm air to evaporate the organic solvents. To the resultant mixture distilled water (10 ml) was added and heated (40-45°C) for 30 min to produce the slurry. An agar solution (1.0 g in 15 ml of distilled water) was heated (50-55°C) and mixed well with the cellulose/compound slurry. The resultant mixture was compressed using a pellet block to prepare approximately five pellets (thickness: 1 cm; diameter: 2.5 cm). Pellets for the control experiment were prepared in the same manner without the test compound. Each pellet (five with the test compound and two without) was transferred to a petri dish and *Glyptotermes dilatatus* (10 each), which had been starved for 24 h, were introduced. Mortality counts were recorded after 5 d from the start of the experiment and were continued for 30 d. Kampferol was used as the positive control, which had 100% mortality in 23 d under the above conditions.

**RESULTS AND DISCUSSION**

*L. sipmanianum* described in this study was collected from an exposed and well-lit, siliceous bed-rock in Beragala (elevation 1077 m above sea-level), Central Province, Sri Lanka. It is a Leprose (loose powdery surface without a cortex) lichen. The thallus contains short distinct lobes with narrow upturned margins and deep cream in colour. *L. sipmanianum* has previously been reported from South Africa, Columbia and Brazil\textsuperscript{15} and the report of it from Beragala is a new record to the entire Asian region.\textsuperscript{16} Previously, we had reported on the presence of *L. sipmanianum* lichen compounds namely, atranorin 1, β-sitosterol 2, (+)-usnic acid 3, zeorin 4 and tritetacontylpentanoate 5 in the butterfly *Talicada nyseus* which is found in the vicinity of the lichen in Beragala.\textsuperscript{17} The butterfly, whose natural food source is *Bryophyllum calycinum* (Sinhala: akkapana; Tamil: *malai-kalli*), found in Beragala was very likely using the lichen as an alternative food source. Although moths of the family Arctiidae are known to be lichen feeders, that there are no records of a Lycaenid butterfly feeding on lichens, makes the association between *L. sipmanianum* and *T. nyseus* unique. In this study, we report on the isolation and structure elucidation of compounds 1-5 and 3β-acetoxyfern-9(11)-ene 6 from the methanol extract of *L. sipmanianum*, and glyceryl trilinolate 7 and 3, 6-dimethyl-2-hydroxy-4-methoxybenzoic acid 8 directly from the lichen powder.

*L. atrotomentosa* was collected from partly shaded, overhung rock surfaces and on soil sheltered from rain in Ramboda (elevation 1700 m above sea-level), Central Province.\textsuperscript{17} It is a Crustose (crust like, thallus stretching over and firmly fixed to the substratum by the whole lower surface and generally lacking of rhizines) type lichen. The thallus is pale blue to grey in colour. There are no previous reports on the chemistry of this species. Chromatography of the lichen powder, in addition to compounds 1-4, yielded methyl-b-orcinolcarboxylate 9.

The HREIMS of the compound 5 gave a peak at \( m/z \) 704.7365, suggesting that the molecular formula was \( C_{40}H_{36}O_8 \). The IR spectrum showed a peak at 1730 cm\(^{-1}\) indicating that this molecule has an ester group which accounted for the one degree of unsaturation suggested by the molecular formula.

The 'H NMR spectrum showed five peaks: a pair of overlapping triplets at \( d 0.84 \) and 0.88 (6H, \( j \sim 5.1 \text{ Hz} \), H5) for the two terminal Me groups; an intense broad multiplet at \( d 1.25 \) (8H, \( j = 10.5 \text{ Hz} \), 2H) representing a CH\(_2\) group which was attached to a CH\(_3\) group. The COSY (Correlation Spectroscopy) spectrum showed a correlation between the proton at \( d 4.05 \) (CH\(_2\)O) and the methylene protons at \( d 1.25 \). The proton at \( d 1.25 \) further showed correlation between the methyl protons at \( d 0.84 \). The foregoing data indicated that compound 5 was a long chain fatty acid ester with both the acid and the alcohol chains ending in a methyl group.

The mass spectral fragments at \( m/z \) 57 and 85 (C\(_4\)H\(_{10}\) and C\(_{49}\)H\(_{44}\)CO) and 620 indicated a butyl group attached to the carbonyl group of an ester and C\(_{56}\) alcohol portion of the molecule, respectively.
Base hydrolysis of the compound gave tritetracontanol whose $^1$H NMR spectrum showed a peak at δ 4.05 (CH$_2$O), which was shifted to downfield compared to the same peak in the starting ester (δ 4.25) confirming that the C$_4$ chain belongs to the alcoholic part of the ester. The $^1$H NMR, COSY and the mass spectral data taken together to confirm this compound 5 as tritetracontylpentanoate. To the best of our knowledge this is a new natural product.

HREIMS of the compound 6 gave a molecular ion at 468.3967 suggesting an elemental composition of C$_{31}$H$_{52}$O$_2$ which was confirmed by $^{13}$C NMR (32 peaks) and DEPT (9 x CH$_3$, 9 x CH$_2$, 7 x CH and 5 quaternary carbons) analysis.

The $^1$H NMR spectrum of compound 6 indicated that it was a triterpenoid with oxygenation in ring A at C-3; it showed six methyl singlets at δ 0.82 (Me-28), 0.86 (Me-29), 0.87 (Me-30), 0.89 (Me-31), 0.91 (Me-32), and 0.94 (Me-33).

Figure 1: Compounds contained in the chemical extract of L.sipmanianum and L.atrotomentosa.
Figure 2: HMB Correlations of 3b-acetoxyfern-9(11)-ene 6.

In the mass spectrum, a peak at m/z 453 was observed, due to the loss of Me group (15 mass units) from the m/z 468 (M'). The base peak at m/z 43 (100) due to a C,H, fragment confirmed the presence of isopropyl side chain.

In order to confirm the identity of this compound as 3b-acetoxyfern-9(11)-ene 6, the compound was hydrolysed with 6% methanolic KOH. The 1H NMR spectrum of the crude hydrolysed product fern-9(11)-ene-3b-ol 6' showed six singlets at d 0.80 (Me-28), 0.82 (Me-30), 0.91 (Me-24), 0.93 (Me-29), 0.97 (Me-27), 1.09 (Me-25) for the methyl groups, a carbinol methine proton at d 3.19 (f = 5.1, 10.35 Hz, H-3), an olefinic proton at d 5.53 (f = 3.8, 8.1 Hz, H-11) and a series of unresolved peaks from d 1.20 to 1.70 (14H). When the 1H NMR data for the two compounds 3b-acetoxyfern-9(11)-ene 6 and fern-9(11)-ene-3b-ol 6' were compared, the multiplicity and the chemical shift value for the olefinic proton was unchanged while the signal for the carbinol methine proton at C-3 moved upfield from d 4.45 to 3.19 (reported value d 3.20). The 1H NMR spectral data of this compound was identical to those reported for fern-9(11)-ene-3b-ol.5 Thus, compound 6 was confirmed as 3b-acetoxyfern-9(11)-ene.3 3b-Acetoxyfern-9(11)-ene 6 and its alcohol have been previously isolated from the plant Rhododendron macrocalplus but this is the first report of compound 6 (containing oxygenation at C-3) from a lichen.6 However, in the related fern-9(11)-ene-ols isolated from lichens, hydroxyl functionality occurs at C-12 or C-19.6

Spectral data (1H NMR, 13C NMR and mass) and physical data (m.p., TLC and co-TLC) were used in the identification of atranorin 1, bistosterol 2, (+)-usnic acid 3, zeorin 4, glyceryl trilinolate 7, 3, 6-dimethyl-2-hydroxy-4-methoxybenzoic acid 8 and methyl-b-orcinolcarboxylate 9.
3,6-Dimethyl-2-hydroxy-4-methoxybenzoic acid 8 at 10 ppm showed moderate mosquito larvicidal activity (90 % moribund after 24 hours) against the second instar larvae of *Aedes aegypti*, which is a major vector of dengue fever. The well-known antimicrobial compound known as (+)-usnic acid 4 showed significant antitermite activity (80 % mortality at 10 mg) against *Glyptotermes dilatatus*, which is a major live-wood attacking termite at low elevations.

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