Recent Strategies in Transition-Metal-Catalyzed Sequential C–H Activation/Annulation for One-Step Construction of Functionalized Indazole Derivatives

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Abstract: Designing new synthetic strategies for indazoles is a prominent topic in contemporary research. The transition-metal-catalyzed C–H activation/annulation sequence has arisen as a favorable tool to construct functionalized indazole derivatives with improved tolerance in medicinal applications, functional flexibility, and structural complexity. In the current review article, we aim to outline and summarize the most common synthetic protocols to use in the synthesis of target indazoles via a transition-metal-catalyzed C–H activation/annulation sequence for the one-step synthesis of functionalized indazole derivatives. We categorized the text according to the metal salts used in the reactions. Some metal salts were used as catalysts, and others may have been used as oxidants and/or for the activation of precatalysts. The roles of some metal salts in the corresponding reaction mechanisms have not been identified. It can be expected that the current synopsis will provide accessible practical guidance to colleagues interested in the subject.

Keywords: C–H functionalization; cyclization; indazole products; transition metals catalysis

1. Introduction

Nitrogen-based heterocyclic systems are commonly found in pharmaceutical agents and natural products, and they have been intensively explored as new bioactive products [1–10]. Among them, the indazoles [11–16] are favored by synthetic and medicinal chemists, as evidenced by their widespread abundance in pharmaceuticals and natural products. In fact, products bearing the indazole segment have been found to show diverse biological activities, including analgesic, anticancer, anesthetic, anti-inflammatory, antifungal, anti-HIV, antiarrhythmic, and anti-emetic properties and high binding affinity for estrogen receptors (Figure 1) [11,17–24]. Pazopanib with an indazole scaffold has been proven to be a potent and selective multitargeted receptor tyrosine kinase inhibitor that inhibits tumor growth [25]. Moreover, derivatives of indazole have been used as electronically active and agricultural materials [26,27]. Indazoles are found in nature as well, and the structures of three natural products containing the indazole scaffold, named nigellicine, nigellicidine, and nigeglanine, are illustrated in Figure 1 [11,28–31].

Notably, the known synthetic procedures toward indazoles may suffer from multistep routes, severe conditions, and relatively low substrate diversity [32–34]. Therefore, considerable attempts have been devoted to constructing these biologically critical scaffolds more efficiently, although the exploration of novel atom-economical strategies to produce these unique scaffolds is still a challenging topic. Recently, a transition-metal-catalyzed C–H bond activation and cyclization sequence [35–40] has received significant attention to provide these structures.
Transition-metal-catalyzed C–H activation [41] and annulation sequences [37,42–48] are powerful and reliable tools for the formation of complex molecular structures in an efficient, economical, practicable, and straightforward manner [37,49]. These synthetic strategies provide highly atom- and step-economic transformations for the development of efficient synthetic procedures with high functional-group tolerance while avoiding the excessive prefunctionalization of reactive centers. One major issue for these protocols is the intrinsic inferior reactivity of C–H bonds, requiring harsh conditions to convert substrates to the target products in satisfactory yields and resulting in low selectivities [50]. To solve this, the C–H activation of inactive arenes via a pendent chelating activator is an efficient route that is additionally incurring site selectivity in C–H activation. Metal salts or metal complexes have commonly been applied for these transformations. Thus, transition-metal-catalyzed sequential C–H activation/annulation reactions of suitable substrates with a variety of coupling partners have been applied to assemble complex indazole-based architectures.

In 2018, a general review appeared on recent advances in various methods for the synthesis of indazole derivatives, focusing on their biological activities [12]. Meanwhile, Li et al. outlined the anticancer activity of indazole derivatives [51] while also summarizing the synthetic protocols and structure–activity relationships of target products [51]. In 2021, Babu et al. comprehensively covered the recent developments in the transition-metal-catalyzed synthesis of indazoles [52]. An overview of the development of novel and green electrochemical approaches in the functionalization of indazole derivatives was reported by Hajra in 2021 [53]. Moreover, a review article for approved marketed drugs containing indazole scaffolds as valid preclinical/clinical drug compounds was published by Wu et al. in 2021 [54].

Because of the significance of indazole heterocyclic systems as well as the rapid development of strategies based on transition-metal-catalyzed sequential C–H activation/annulation for the one-step synthesis of functionalized indazole derivatives, a dedicated comprehensive overview would be timely and beneficial for future drug discovery.

Consequently, we aim to present a review arranged according to the various sorts of metal salts or metal complexes applied in such transition-metal-catalyzed C–H activation and annulation sequences. The purpose of the current overview is to report the recent exploration in this area based on different transition metal catalysts derived from rhodium, cobalt, palladium, rhenium, and copper. While covering the subject, a variety of examples and selected mechanisms of reactions are discussed.
2. Different Synthesis Routes for the Construction of Indazoles via Transition-Metal-Catalyzed Sequential C–H Activation/Annulation

2.1. Synthesis of Indazoles Using Rhodium and Copper Salts

In 2016, a facile and efficient access to 1H-indazoles 3 was established through Rh(III)/Cu(II)-catalyzed sequential C–H bond activation and intramolecular cascade annulation. The reaction occurs at 80 °C within 24 h in PhCF₃ as a solvent. A comprehensive examination of this process was conducted using ethyl benzimidates 1 and nitrosobenzenes 2. A control experiment without adding Rh or Cu catalysts was then run to demonstrate that this transformation could not proceed without either one of these catalysts. The authors proposed a significant facilitation role of the bridging acetate ligand in the Rh₂(OAc)₄ structure for the C–H activation. Benzimidate scaffolds 1 with both electron-withdrawing and electron-donating functional groups as well as halogens all worked well to afford the desired products with moderate to high yields. Furthermore, imidate substrates 1 that bear other alkyl esters as well as aryl substitution were transformed into the corresponding products 3 smoothly. Subsequently, a diverse range of nitrosobenzenes 2 with electron-donating and -withdrawing groups at different positions of the aryl ring were also proven to be viable substrates. Significantly, this transformation features satisfactory functional-group tolerance with good to high yields (Scheme 1) [55].

![Scheme 1. A synthetic route for the preparation of 1H-indazole derivatives 3 from the reaction between imidates 1 and nitrosobenzenes 2 in the presence of a rhodium/copper catalyst.](image)

The mechanism for the formation of 1H-indazoles 3 through the reaction of imidates 1 with nitrosobenzenes 2 catalyzed by a rhodium/copper catalyst is given in Scheme 2. A rhodacycle 5 is generated by the coordination of an imidate 1 to a catalyst 4 and C–H activation. Subsequently, the migratory insertion of the Rh–C bond into the N=O group yields a six-membered rhodacycle 6. In the next step, the protonolysis of the intermediate 6 delivers a hydroxylamine 7 along with a regenerated Rh(III) catalyst 4. In another cycle, N–O oxidative addition is carried out using the Cu(I) catalyst to deliver a Cu(III) organocupracycle 8. Finally, the dehydration of 8 and the reductive elimination of 9 take place to form the N–N bond [55]. Although the authors reported intermediates 8 and 9, it seems Cu(III) would involve a chloride or another negatively charged ligand attached to copper in intermediates 8 and 9.

A Rh(III)-mediated substrate-controlled conversion of azobenzenes 10 and alkenes 11 to indazoles 12 was described via C–H functionalization and cyclization (Scheme 3). First, 5 mol% of [Cp*RhCl₂]₂ and 200 mol% of Cu(OAc)₂ as oxidant and DCE as solvent were utilized to transfer azobenzenes 10 to indazoles 12 under nitrogen atmosphere in good to excellent yields. To highlight the importance of having both metals present, no desired compound was formed when either the Rh(III) as catalyst or Cu(II) as oxidant were removed. The scope of this C–H functionalization/cyclization reaction with regard to both azobenzenes 10 and alkenes 11 was screened. Acrylates 11 with different substitutions efficiently proceeded to afford final products in satisfactory yields. However, phenyl vinyl sulfone and dimethyl vinyl phosphonate did not react with its azobenzyme partner under optimized conditions. Azobenzene substrates possessing both electron-donating and electron-withdrawing functional groups were well-tolerated, as evidenced by the isolation of the desired products in moderate to good yields. The product with a stronger
The subsequent utilization of the present strategy could be a rapid and straightforward approach for the synthesis of new functional and biologically active molecules [56].

Scheme 2. A rational mechanism for the synthesis of 1H-indazole derivatives 3 from the reaction between imidates 1 and nitrosobenzenes 2 in the presence of a rhodium/copper catalyst.

Scheme 3. A synthetic route for the transformation of azobenzene substrates 10 to indazole derivatives 11 in the presence of a Rh(III) catalyst.

The mechanism for the synthesis of indazoles 12 from azobenzenes 10 is exhibited in Scheme 4, which involves coordination, C–H activation, alkene coordination and insertion,
β-hydride elimination, the insertion of a C=C bond into the Rh-N bond, and then aromatization to produce indazoles. It was proposed that copper acetate plays its role in the step of the β-hydride elimination to give indazoles [56].

Scheme 4. A possible mechanism for the transformation of azobenzene substrates 10 to indazole derivatives 12 in the presence of a Rh(III) catalyst.

A novel methodology for the synthesis of 2,3-dihydro-1H-indazoles 19 via the oxidative olefination of 1,2-disubstituted aryldrazines 18 with alkenes 11 through C(sp²)-H bond functionalization pursued by an intramolecular aza-Michael transformation was reported by Kim et al. A series of highly substituted 2,3-dihydro-1H-indazoles 19 with different substituents were obtained in low to high yields in the presence of [RhCp*Cl₂] as a catalyst and Cu(OAc)₂ as an oxidant in MeCN under air atmosphere at 100 °C for 20 h in pressure tubes. The substrate scope was subsequently explored under this catalytic system, as illustrated in Scheme 5. Various acrylates 11 and N'-methyl-N'-arylacetoxydrazides 18 with both electron-donating and -withdrawing substituents at the para-site of the aromatic ring were explored, giving the desired products 19 via oxidative olefination and the subsequent intramolecular cyclization in moderate to good yields. The functional-group tolerance, especially to the bromo and chloro groups, provides a versatile synthetic protocol for the further functionalization of the 2,3-dihydro-1H-indazoles 19. Interestingly, when meta-substituted aryldrazine drazides 19 were used as the reaction substrates, Rh(III)-catalyzed oxidative coupling preferentially occurred at the less hindered site, providing the corresponding products 19 in a regioselective manner. Moreover, ortho-substituted compounds 18 afforded the desired products with slightly decreased eectivity, which is presumably attributed to the steric effect of a substituent at the ortho-position. The corresponding products 19 were further produced in low to high yields by the utilization of diverse olefins 11. For acrylates containing -CO₂Me, -CO₂nBu, -CO₂Bu, -Bn, and -CO₂Ph as EWGs and acrylamide (containing -CONMe₂ as an EWG), 2,3-dihydro-1H-indazoles 19 were generated in high yields, while acrylonitrile 11 (containing -CN as an EWG) and but-3-en-2-one 11 (containing -COMe as an EWG) were reacted with N'-methyl-N'-phenylacetohydrazide with significantly decreased yield under the standard conditions. In the case of acrylate containing an estrogen scaffold, the desired product 19 was obtained in the reaction with N'-methyl-N'-phenylacetohydrazide. The bis-indazole 19aa was then achieved in 63% yield by increasing the amounts of hydrazide, rhodium as a catalyst, and copper as an oxidant [57].
of arylhydrazines 19, the structurally diverse alkenes 19 reacted with N'-methyl-N'-phenylacetohydrazide to afford indazole derivatives 19 in good to high yields. A series of the N'-alkyl-N'-arylacetohydrazide derivatives 18 can smoothly be converted into corresponding products 19 with moderate to good yields as well. Disubstituted N'-alkyl-N'-arylacetohydrazides and N'-alkyl-N'-arylacetohydrazides 18 bearing both electron-rich and electron-deficient groups including methyl, methoxy, fluoro, chloro, and bromo on the aryl ring all displayed moderate to good reactivities in the sequence smoothly furnishing the desired indazoles 19. In the case of meta-substituted N'-alkyl-N'-arylacetohydrazides, the reaction proceeded successfully at the less steric side of the arenes. Moreover, 18 containing an ortho-substituent also participated in this reaction to afford the corresponding product, although the yield greatly decreased. By having ethyl and n-butyl instead of methyl at the N-atom, the corresponding products 19 were obtained in moderate yields. This transformation provided a practical procedure to achieve useful target products 19 through C–H bond activation, featuring good to excellent yields with excellent regioselectivity, high atom economy, and versatile derivatization [58].
A catalytic reaction of 1-alkynylcyclobutanols 20 with pyrazolidinones 21 toward pyrazolo[1,2-a] indazoles 22 bearing a quaternary carbon was developed by Yu et al. in 2020. The reaction proceeds through Rh(III)-catalyzed sequential C–H/N-H activation and ring opening, followed by cyclization transformation in a one-pot process. This protocol provides an efficient and atom-economical approach for the direct synthesis of pyrazolo[1,2-a] indazoles 22 in high yields with exclusive regioselectivity under the optimized conditions (using catalytic amounts of [Cp*RhCl₂] and Cul in the presence of Na₂CO₃ in DCE at 50 °C within 12 h under Ar atmosphere) (Scheme 7). First, the reaction between 1-phenylpyrazolidin-2-one 20 and 1-alkynylcyclobutanols 21 bearing several electron-rich and electron-poor substituents at different positions on the phenyl ring was explored, affording the desired products in good to high yields. When the standard reaction conditions were applied to 1-alkynylcyclobutanols 21 with para- and meta-substituted phenyl groups, high yields of products were obtained. The usage of 1-(phenylethylnyl) cyclobutanols 20 containing an electron-poor or electron-rich group, including halide or methoxy at the ortho-position of the benzene had little effect, and pyrazolo[1,2-a] indazole derivatives were achieved with good yields. Even the substrates substituted with heterocyclic naphthyl, thienyl, and pyridyl groups or alkyl groups could be used in the coupling reaction with satisfactory yields. Surprisingly, by the replacement of the cyclobutanol moiety of 1-(phenylethylnyl) cyclobutanol with cyclohexanol, a seven-membered ring product 22a was obtained with good yield. In order to better understand the substrate scope of this [4 + 1] cyclization and ring opening, a series of N-arylpyrazolidinones 20 were further tested under standard conditions. The coupling transformations proceeded successfully to form the desired indazoles 22 in moderate and high yields by the introduction of substituents such as -Cl, -Br, -CN, -Me, or -OMe at the para-site and -Cl, -Br, -Me, -OMe at the meta-site of the aryl ring. The m-OMe-substituted pyrazolidinone 20 displayed slightly lower regioselectivity for this reaction. Two pyrazolidinones substituted at the 3- and 5-positions were examined, and these delivered the desired products in reasonable yields. However, the [4 + 1] annulation and ring opening failed to form the corresponding products 22 when the ortho-substituted pyrazolidinones 20 possessing steric hindrance were exploited. No product was formed by the replacement of the phenyl moiety of pyrazolidinone 20 with a pyridyl group. The current procedure showed high functional-group tolerance and great efficiency, providing a variety of corresponding compounds 22 in moderate to good yields under mild conditions [59].

Scheme 7. A synthetic route for the preparation of pyrazolo[1,2-a] indazoles 22 containing a quaternary carbon in the presence of a Rh(III) catalyst.
2.2. Synthesis of Indazoles Using Ruthenium Salt in the Presence of Sodium Acetate

Ru(II)-catalyzed tandem ortho-carbonylation/amidation/cyclization of 2-aryl-2,3-dihydrophthalazine-1,4-diones 23 has been developed towards the direct assembly of substituted indazolo[1,2-b]phthalazine-triones 25. The reaction conditions involve [RuCl₂(p-cymene)]₂ (5 mol%) and NaOAc (50 mol%) in DCE as a solvent at 40 °C within 4 h under N₂ atmosphere (Scheme 8). The scope and limitations of this tandem ortho-carbonylation/amidation/cyclization strategy with respect to 2-aryl-2,3-dihydrophthalazine-1,4-diones 23 have been explored. In general, good to high yields of the corresponding products were achieved by using meta- and para-substituted substrates 23 with both electron-rich, electron-poor, and halogen functional groups. However, the substrates 23 substituted at the ortho site underwent this tandem reaction only under harsh reaction conditions (DCE at 120 °C for 6 h) to afford the products in moderate to high yields. Plus, disubstituted substrates 23 with substituents located at the meta- and para-sites were amenable to give the desired products in DCE at 40 °C within 4 h, although the substrate with substituents located at the ortho- and para-sites gave the targeted product in DCE only after heating at 120 °C for 6 h. In the case of 6/7-substituted phthalazine derivatives (phthalazine scaffold with 6/7-Me, 6/7-OMe, 6/7-t-butyl, 6/7-Br, 6/7-F, and 6/7-Cl), mostly inseparable mixtures of the corresponding products 25a–f were obtained. Indeed, a mixture of products was obtained, as a mixture of the starting materials was applied in the reaction.

Phenyldrazines containing 5-methyl, 5-methoxy, 5-t-butyl, 5-bromo, and 4-fluorophthalic anhydrides provided inseparable mixtures of 6/7-methyl, 6/7-methoxy, 6/7-t-butyl, 6/7-bromo, and 5/8-fluoro-2-phenyl-2,3-dihydrophthalazine-1,4-diones 25a–e, while phenylhydrazine condensation products with 4-chlorophthalic anhydrides provided 5/8-chloro-2-phenyl-2,3-dihydrophthalazine-1,4-dione 25f as a single regioisomer. The amino, nitro, trifluoromethyl, or cyano functional groups on aryl/phthalazine moieties did not work under the optimized reaction conditions [21].

A plausible mechanism for this reaction is illustrated in Scheme 9. Initially, an active catalyst was formed in situ from [RuCl₂(p-cymene)]₂ using NaOAc. The monomeric [Ru(II)OAc₂(p-cymene)] species transferred to 26 via a ligand exchange with N-H of 23, followed by C–H metalation to deliver a five-membered ruthenacyclic complex 27. The
catalytic center was then coordinated by the C=N double bond of aryl isocyanate 24. Next, the migratory insertion of –C=N into a Ru-Ar bond gives a seven-membered ruthenacyclic complex 29. The protonolysis of the two nitrogens in intermediate 29, followed by an intramolecular nucleophilic substitution, provided a tetracyclic carbocyclized product 25 [21].

Scheme 9. A rational mechanism for the synthesis of indazolo[1,2-b]phthalazine-trione derivatives 25 from the reaction between 2-aryl-2,3-dihydrophthalazine-1,4-diones 23 and isocyanates 24 under ruthenium catalysis.

2.3. Synthesis of Indazoles Using Rhodium Complexes and Silver Salts

An interesting methodology for the synthesis of 3-acyl-2H-indazoles 33 was performed using a Rh(III)-catalyzed tandem C–H alkylation/intramolecular decarboxylative cyclization of aza compounds 31 with diazoesters 32. The combination of catalytic amounts of [Cp*RhCl2]<sub>2</sub>/AgSbF<sub>6</sub> in the presence of PivOH in DCE/dioxane plays a crucial role to afford the corresponding products at 130 °C within 24 h. In this research, aza scaffold 31 was introduced as an incorporated directing group for regio- and chemoselective [4 + 1] cyclization, which could be promoted by diverse symmetrical and unsymmetrical dia- and mono-aryldiazene oxides. The feasibility of the coupling of symmetrical and unsymmetrical dia- and mono-aryldiazene oxides with diazoesters 32 was explored under standard conditions. First, several diazoesters 32 were investigated, affording H-indazoles. While the disopropyl 2-diazo malonate led to a high yield, diazoester containing t-Bu led to a moderate yield, assumably owing to the easy hydrolysis of this functional group under the optimized reaction conditions. Moreover, α-diazo-β-keto esters participated in this annulation to the corresponding products in good to high yields. Generally, a wide range of functional-group tolerance for both α-diazo-β-alkyl keto esters and α-diazo-β-(hetero)aryl esters highlights this strategy for the regio- and chemoselective synthesis of 2H-indazoles. However, when diazoesters with N-heterocycles, including methyl 2-diazo-3-oxo-3-(pyridin-2-yl)propanoate, methyl 2-diazo-2-(pyridin-2-yl)acetate, and methyl 2-(benzol[d]thiazol-2-yl)-2-diazoacetate, were used under optimized reaction conditions, no product was formed. The diazo reagents including Et<sub>2</sub>O-C-CHN<sub>2</sub>, Et<sub>2</sub>O-C-C(CH<sub>3</sub>)<sub>3</sub>N<sub>2</sub>, Et<sub>2</sub>O-C-CPH<sub>2</sub>, and Et<sub>2</sub>O-C-C(CF<sub>3</sub>)N<sub>2</sub> also did not work with this catalytic system. While symmetrical azoxybenzenes substituted at the ortho, meta, and para-positions of aryl rings could be applicable in this reaction, azoxybenzenes substituted at the meta-position of aryl rings delivered a fairly low yield, conceivably
because of steric congestion. The azoxybenzenes substituted at the ortho-position could be exploited as coupling partners, with coupling occurring exclusively in the sterically less hindered site. Some more complex molecules 33a–e are shown in Scheme 10. The advantages and benefits of this transformation are regioselectivity for unsymmetrical azoxybenzenes and the compatibility of monoaryldiazene oxides [60].

![Scheme 9. A synthetic route for the preparation of 2H-indazoles derivatives 33 via the regio- and chemoselective [4 + 1] annulation of azoxy compounds 31 with diazoesters 32 under rhodium catalysis.](image)

Scheme 10. A synthetic route for the preparation of 2H-indazoles derivatives 33 via the regio- and chemoselective [4 + 1] annulation of azoxy compounds 31 with diazoesters 32 under rhodium catalysis.

The rhodium(III)-catalyzed synthesis of indazole derivatives 36 has been realized through an intermolecular C–H amination and N–N bond formation starting from readily available ketoxime ethers 34 and 4-toluenesulfonamides 35a (Scheme 11). All reactions were carried out using 2.5 mol% of [Cp*RhCl_2]_2 as a catalyst, 0.6 mmol PhI(OAc)_2, and 10 mol% of AgSbF_6 in trifluoroethanol (TFE) at 90 °C for 24 h. First, the functional-group tolerance of the ketoxime derivatives 34 was explored under optimized reaction conditions. Although the reaction of meta- or para-nitro-substituted acetonaphone oxime derivatives was successfully carried out to afford moderate to good yields of desired products, the 2-nitrobenzaldehyde oxime methyl ether only gave the corresponding indazole 36 in a moderate yield due to steric effects on the aryl ring. The unsubstituted acetonaphone oxime derivatives were all suitable for this system, leading to desired indazoles in satisfactory yields. In the next step, both p-ester- and cyano-substituted acetonaphone oxime ethers were proven to be appropriate substrates for this transformation as well. The acetonaphone oxime derivative exhibited less efficacy in this oxidative annulation, affording the expected indazole in a 15% yield. Afterward, the authors investigated more amides to address the low reactivity of 4-toluenesulfonamide 35a in this reaction. The results showed that phenylsulfonamides 35b containing electron-deficient substituents produced better yields of the target indazoles 36. Several substituted acetonaphone oxime ethers 34 derived from propiophenone, n-butyrophenone, cyclopropyl phenyl ketone, and diphenylketone exhibited good compatibility for the reaction with 4-nitrobenzenesulfonamide 35b. Notably, substituents such as F, Br, I, and CF_3 on the aromatic ring of the acetonaphone oxime ethers all survived the reaction conditions, affording the desired products 36 in good yields. To sum up, a broad range of substituents were possible with this catalytic system, producing different functionalized indazoles with acceptable yields [61].
A possible mechanism for the synthesis of indazole derivatives 36 via the oxidative annulation of ketoxime methyl ethers 34 with sulfonamide 35a–b under rhodium catalysis.

The mechanism involves (Scheme 12) the electrophilic activation of Rh by chloride removal with the silver salt, the coordination of ketoxime ether 34 with an active catalyst, and C–H activation, respectively, to form the complex 37. This complex is oxidized in the presence of in situ-generated iodonium 38 to give the complex 39. In the next step, migratory insertion and releasing [Cp*Rh(SbF₆)₂] as the active complex occurs to form 41. Finally, target product 36 is obtained via the reaction with iodobenzene diacetate [61].

Scheme 11. A synthetic route for the preparation of indazoles derivatives 36 via the oxidative annulation of ketoxime methyl ethers 34 with sulfonamide 35a–b under rhodium catalysis.

Scheme 12. A possible mechanism for the synthesis of indazole derivatives 36 via the oxidative annulation of ketoxime ethers 34 with sulfonamide 35 under rhodium catalysis.

2.4. Synthesis of Indazoles Using Rhodium and Silver Salts in the Presence of Lithium Acetate

The synthesis of C3-unsubstituted and C3-trifluoromethylated (2H)-indazoles starting from azobenzenes 10 and paraformaldehyde 42 as a useful C1-feedstock under rhodium catalysis was reported by Kim et al. in 2019. The strategy was also successfully expanded to C–H activation/annulation reactions of azobenzenes 10 with trifluoroacetaldehyde hemiacetal 44. The optimized conditions for the synthesis of (2H)-indazoles 43 and 45 from azobenzenes 10 included 2.5 mol% of [RhCp*Cl₂], 10 mol% of AgSbF₆, 30 mol% of LiOAc,
and 30 mol% of Ag$_2$CO$_3$ in DCE as a solvent at 120 °C for 12 h (or 150 °C for 20 h) under air atmosphere in sealed reaction tubes. LiOAc and AgSbF$_6$ were used to activate the Rh catalyst by forming [Cp*Rh(III)(OAc)]SbF$_6$. Various symmetrical ortho-, meta-, and para-disubstituted azobenzenes underwent coupling, with paraformaldehyde 42 showing good functional-group tolerance. While symmetrical 2,5-disubstituted azobenzenes (5-fluoro-2-methyl azobenzene and 5-chloro-2-methyl azobenzene) were well-tolerated and the desired products were obtained in 58% and 12% yields, respectively, sterically congested 2,5-dimethyl azobenzene failed to afford the corresponding product. In the next step, several unsymmetrical azobenzenes were screened as substrates, and the products 43a–d and 43a’–43d’ were obtained (Scheme 13). Notably, the steric environment of the azobenzene orients the formation of desired products 43e–f. The substrate scope of this reaction was further expanded to trifluoroacetaldehyde ethyl hemiacetal 44 to produce a range of C3-CF$_3$-substituted (2H)-indazoles 45 in moderate to high yields. This conversion efficiently afforded several C3-unsubstituted and C3-trifluoromethylated (2H)-indazoles 43 and 45, which are important molecules in organic chemistry and pharmaceutical sciences. The practicability of this approach is highlighted by its chemoselectivity, functional-group tolerance, and wide substrate scope [62].

Scheme 13. A synthetic route for the preparation of 1H-indazole derivatives 43 and 45 from azobenzenes 10 using paraformaldehyde 42 as a one-carbon synthon or trifluoroacetaldehyde ethyl hemiacetal 44.

2.5. Synthesis of Indazoles Using Rhodium and Silver Salts in the Presence of Sodium Acetate

Yu’s group developed an efficient and novel Rh(III)-catalyzed annihilation transformation between phthalazinones/pyridazinones 23 and 46 and different allenes 47 into indazole derivatives 48 containing a quaternary carbon (Scheme 14). A C–H activation and olefin insertion sequence, followed by β-hydride elimination and intramolecular annulation occurs to produce indazole derivatives 48 using a catalytic amount of [Cp*RhCl$_2$]$_2$ in the presence of AgOAc in MeCN at 100 °C within 12 h under air atmosphere. The active Rh catalyst is formed in situ by the ion exchange of chloride to acetate involving [Cp*RhCl$_2$]$_2$ and AgOAc/NaOAc. In other words, acetate ions replace chloride ligands to give [Cp*Rh(OAc)$_2$]. N-Aryl phthalazinone and pyridazinone substrates bearing a
range of electron-rich electron-poor substituents at different positions could deliver the corresponding products in satisfactory yields. The [4 + 1] cyclization of the substrate substituted by methyl at the ortho-position of the N-aryl affords the desired product in only a 26% yield because of steric hindrance. On the other hand, the reaction of the N-aryl substrate with methyl at the meta-position demonstrated remarkable reactivity and excellent chemoselectivity, while the corresponding meta-methoxy analogs gave an isomeric mixture in a 9:1 ratio. Para-substituted N-aryl phthalazinone substrates 23 with a variety of electron-rich and electron-poor functional groups can be easily transformed into the desired indazoles 48 in good to high yields under standard reaction conditions. It is worth noting that the substrate substituted by two methyl groups at the 3 and 5 positions of the N-aryl segment worked well to give the corresponding product in a good yield. The products 48a and 48b could be formed with moderate selectivity (1:1). The products 48c and 48d were also obtained under standard conditions. The target compounds 48e–m were also achieved via this [4 + 1] cyclization in good yields. The substrate scope of a variety of allenes 47 substituted by several electron-donating or electron-deficient groups at different positions gave the corresponding products in acceptable to high yields. 1-Naphthyl- or 2-naphthyl-group-substituted allenes and aliphatic allene were also well-tolerated under these reaction conditions, thus producing the corresponding indazoles 48n–q in good to high yields. Notably, a long-chain aryl-substituted substrate yielded the desired product 48q with good selectivity (E/Z = 9:1) in a high yield. Finally, 1,1,4-trisubstituted and simple aliphatic allenes containing an ester group did not work under optimized reaction conditions. The approach successfully tolerates different substituents on the starting phthalazinones/pyridazinones and allenes and features practicability, high atom efficiency, and high Z-selectivity [63].

The synthesis of C3-hydroxymethylated (2H)-indazoles 50 can be achieved through a subsequential Rh(III)-catalyzed C–H functionalization/intramolecular cyclization between azobenzenes 10 and vinylene carbonate 49 using catalytic amounts of [RhCp*Cl2]2, AgSbF6, and NaOAc in DCE at 80 °C within 14 h under air atmosphere in pressure tubes (Scheme 15). The electronic properties of the azobenzene rings can orient towards the construction of (2H)-indazoles 50 or the isomeric dihydrocinnolin-4-ones 51. It is worth noting that the vinylene carbonate substrate plays a significant role in the promotion of the [4 + 1] or [4 + 2] cyclization. The substrate scope and limitations of the azobenzenes 10 were tested under optimized reaction conditions. The coupling transformations were successful for ortho-substituted azobenzenes containing electron-donating substituents, obtaining moderate to high yields of C3-hydroxymethylated (2H)-indazoles. This catalytic system allows moderate functional-group tolerance, particularly for –F, –Cl, and –Br substituents. No corresponding product 50a was formed by the utilization of sterically congested 2,5-dimethyl-substituted azobenzene. Meta- and para-methyl substituted azobenzenes were also applicable using this catalytic system, although the desired products 50b and 50c were obtained with relatively low yields. Moreover, para-OMe azobenzene was proven to be an efficient substrate, producing 50d in good yield. This exploration was further extended to the coupling of unsymmetrical azobenzenes with vinylene carbonate 49. For example, azobenzenes containing electron-donating substituents on both aryl rings afforded almost equimolar ratios of 50e and 50e' under standard conditions. In the case of other unsymmetrical azobenzenes, the transformation mainly occurred on the more electron-donating aromatic ring, as for 50f (50f') and 50g (50g'). Furthermore, it was shown that no C3-hydroxymethylated (2H)-indazole was obtained using azobenzenes containing electron-poor substituents, and only 2,3-dihydrocinnolin-4-ones 51 were formed via [4 + 2] annihilation transformation. When unsubstituted azobenzene was used as a substrate in the reaction, a mixture of both (2H)-indazole and 2,3-dihydrocinnolin-4-one was produced. The notable benefits of the methodology are mild reaction conditions and excellent functional group compatibility, thus furnishing easy access to a series of indazoles of interest in organic chemistry and pharmacy [64].
The C–H bond activation of phthalazinones 23 or pyridazinones 46 and allenes 47 under Rh(III) catalysis.

Scheme 15. A synthetic route for the preparation of 2H-indazole derivatives 50 via the annulation of azobenzenes 10 with vinylene carbonate 49 under Rh(III) catalysis.
The mechanism for the synthesis of (2H)-indazoles 50 via the annulation of azobenzenes 10 with vinylene carbonate 49 under Rh(III) catalysis is illustrated in Scheme 16. First, the active catalyst is formed from [Cp*RhCl₂]₂ and AgSbF₆ in the presence of NaOAc. The C–H bond of azobenzene 10 is activated to deliver a five-membered rhodacycle intermediate 52. Subsequently, complex 52 undergoes olefin coordination, migratory insertion, and protonation to afford the ortho-alkylated compound 55. In the next step, Ag⁺ acts as a Lewis acid to activate the nucleophilic substitution of an azo group at the α-position on the dioxolan-2-one scaffold, followed by an aromatization transformation to produce (2H)-indazole 50 [64].

Scheme 16. A possible mechanism for the synthesis of 2H-indazole derivatives 50 via the annulation of azobenzenes 10 with vinylene carbonate 49 under Rh(III) catalysis.

2.6. Synthesis of Indazoles Using Rhodium and Silver Salts in the Presence of Zinc Acetate

The commercially available [Cp*RhCl₂]₂ has been exploited as a catalyst for the reaction of phenylhydrazines 58 with 1-alkynylcyclobutanols 21 in the presence of SbF₆ and Zn(OAc)₂ in toluene at 50 °C for 24 h under Ar atmosphere (Scheme 17). Apparently, the reaction proceeds via a hydrazine-directed C–H functionalization process. This catalytic system provided an efficient protocol to produce 1H-indazole derivatives 59 via [4 + 1] annulation based on the cleavage of an alkyne bond with good functional group compatibility and moderate to good yields. First, the substrate scope of 1-arylethynyl cyclobutanols 21 was investigated, and all the results showed that 1-arylethynyl cyclobutanol derivatives containing both electron-deficient and electron-rich substituents on the para- or meta-sites of the benzene ring could afford the corresponding indazoles 59 in acceptable yields. It was found that di-fluoro or 2-thienyl substituted substrates could deliver the corresponding indazole in satisfactory yields using AgNO₃ instead of AgSbF₆. In the next step, several arylhydrazines substituted at the position of the aryl-linked nitrogen were screened. While 1-ethyl and 1-benzyl phenylhydrazines delivered the corresponding indazoles 59 in good yields, the N-phenyl substituted substrate did not work in this reaction, presumably because of its higher steric hindrance. Arylhydrazines containing halides such as F, Cl, and Br as well as electron-rich substituents such as Me and OMe provided the desired products in satisfactory yields. Notably, 1-alkyl-1-phenylhydrazines substituted at the ortho-position were compatible under optimized reaction conditions [65].
A plausible mechanism for the synthesis of 1H-indazoles 59 via hydrazine-directed C–H functionalization with 1-alkynylcyclobutanols 58.

The first step of the mechanism comprises the activation of a rhodium catalyst using AgSbF$_6$/Zn(OAc)$_2$. In the case of Zn(OAc)$_2$, apparently the zinc ion itself plays no role, and only the acetate ion activates the Rh catalyst (Scheme 18). Then, the hydrazine group of 58 coordinates to the metal center of this active complex. Next, C(aryl)−H activation, regioselective migratory insertion, and β-carbon elimination happen to form a six-membered rhodacycle intermediate 63. In the next step, an intramolecular nucleophilic attack of phenyl hydrazine nitrogen on rhodium(III) ion affords intermediate 64. Finally, the target molecule 59 is formed via an intramolecular Michael addition, a proton transfer, and releasing a rhodium-carbene 67 [65].

Scheme 17. A synthetic route for the preparation of 1H-indazoles 59 via hydrazine-directed C–H functionalization with 1-alkynylcyclobutanols 58.

2.7. Synthesis of Indazoles Using Rhodium and Silver Salts in the Presence of Magnesium Sulfate

A convenient transformation for the construction of substituted N-aryl-2H-indazole derivatives 71 through the reaction of azobenzenes 10 with aldehydes 70 in a simple, effi-
Whereas the unsymmetrical para-nitro azobenzene provided major product 71a was promoted by 5 mol% (Cp*RhCl)2 and 20 mol% AgSbF6 in dioxane at 80 °C for 24 h. Whereas the unsymmetrical para-nitro azobenzene provided major product 71a by annulation on the more electron-rich phenyl ring of the azobenzene, 4-methoxy-substituted azobenzene produced a mixture of products 71b and 71b' in low selectivity in which a C–H activation/annulation sequence occurred on the less electron-donating phenyl ring of the azobenzene. In contrast to para-nitro substituted azobenzene, the 3-methoxy substituted azobenzene produced a mixture of products indicating that steric effects are important. In the case of hydroxy-derivatives bearing both electron-deficient and electron-donating substituents proved to be suitable in the reaction with hydroxy-substituted azobenzene, the yield of product 71f could be improved by the utilization of MgSO4 as a drying agent and THF as a solvent. Next, aromatic aldehydes bearing both electron-deficient and electron-donating substituents proved to be suitable in the reaction with a 4-hydroxy-3,5-dimethylphenyl azobenzene substrate, leading to good to high yields of products 71h–v. Notably, aliphatic aldehyde provided the desired indazole 71n in a low yield. This reaction offers remarkable advantages, including the simplicity of the reaction, excellent regioselectivity, good functional-group tolerance, and the exploitation of commercially available materials [56].

Scheme 19. A synthetic route for the preparation of indazole derivatives 71 via C–H bond functionalization and cyclative capture.
2.8. Synthesis of Indazoles Using Ruthenium and Copper Salts in the Presence of Potassium Hexafluorophosphate

A Ru-catalyzed cascade reaction between N-aryl phthalazinediones 23 or N-aryl pyridazinediones 46 with acrylates proceeded through a subsequential oxidative alkenylation/intramolecular cyclization in aqueous media as a green solvent (Scheme 20). All reactions proceeded successfully in the presence of [RuCl₂(p-cymene)]₂ (5 mol%) as a catalyst, KPF₆ (10 mol%) as an additive, and Cu(OAc)₂·H₂O (1 eq) as an oxidant to provide moderate to high yields of products. The limitations and diversity of this alkenylation–annulation via C–H bond activation were explored with respect to N-aryl phthalazinedione 23 and N-aryl pyridazinediones 46 derivatives. N-Aryl phthalazinediones 23 and N-aryl pyridazinediones 46 containing electron-deficient or electron-rich substitutions on the aromatic rings afforded the desired indazole derivatives 72 with moderate to excellent yields. The reaction conditions did not work for methyl methacrylate. Moderate yields of desired products 74a and 74b were achieved by using substrates bearing a pyrazolidinone scaffold [67].

Scheme 20. A synthetic route for the preparation of indazole derivatives 72, 73, and 74 via an alkenylation–annulation approach.

Subsequent oxidative vinylolation and annulation are the key steps for this transformation. A possible mechanism for the reaction is shown in Scheme 21. The transformation is initiated by N–H assisted C–H bond ruthenation of compound 23 or 46 to construct a ruthenacycle complex 74. Next, migratory insertion, β-hydride-elimination, and reductive elimination occurred to form alkenylated product 78. Eventually, an intramolecular aza-Michael addition reaction took place to produce the target indazoles 72 or 73 [67].
of the active catalyst to give rhodacyclic complex 72 or 73 via an alkenylation–annulation approach.

2.9. Synthesis of Indazoles Using Rhodium, Copper, and Silver Salts

In 2013, Glorius and co-workers developed a process involving Rh(III)-catalyzed C−H activation/C−N bond formation and Cu-catalyzed N−N bond formation under mild reaction conditions for the synthesis of substituted 1*H*-indazoles 80 using [Cp*RhCl₂]₂, Cu(OAc)₂, and AgSbF₆ catalysts and molecular oxygen as the oxidant (Scheme 22). For the first time, in this study, it was possible to perform the transformation of N-H-imidates 1 to corresponding compounds 80. Diverse arylsulfonyl azide derivatives 79 with 4-Me, 4-OME, 4-NO₂, 4-F, 4-Cl, and 4-Br as substituents as well as alkylsulfonyl azides 79 (R₂ = Bn, Bu) demonstrated moderate to high reactivity in this conversion. Several arylimidates were exploited to explore the reactivity of different substituents in the alkoxy and arene parts. The ethylimidate 1 (R₄ = OEt) revealed a more satisfactory yield than methyl 1 (R₄ = OMe) and isopropyl 1 (R₄ = O-i-Pr) derivatives. Contrary to the electron-donating arylimidates, the electron-withdrawing imidates afforded better yields for this transformation. This cascade reaction is practical, scalable, and green, using O₂ as the stoichiometric oxidant. In addition, only N₂ and H₂O are the byproducts of this reaction. It is worth noting that indazole was formed with a <5% yield when omitting either of the Rh or Cu catalysts for the model reaction, denoting the important role of these two catalysts in the mechanism cycle [68].

Mechanistically, the construction of indazoles 80 comprises the steps shown below in Scheme 23. First, the active catalyst (cationic [Cp*Rh(III)]) is obtained in situ using AgSbF₆ to remove the chloride. Subsequently, coordination to imidate 1 and C−H activation are performed by the active catalyst to give rhodacyclic complex 81. The coordination of azide substrate 79 to the catalyst followed by nitrogen loss and migratory insertion yields Rh(III) amido species 83. The resulting intermediate 83 is protonated to regenerate the active catalyst and provide the amidated compound 84. In the next step, the resulting compound may coordinate to the copper source to form intermediate 85. The authors considered two paths for the continuation of the mechanism. In route A, the Cu(III) complex is transferred to a higher valent Cu(III) complex in the presence of another molecule of Cu(OAc)₂ or O₂. Next, the construction of the N−N bond via reductive elimination formed the target compound 80. In the final step, Cu(II) complex 87 is transformed to Cu(OAc)₂ in the presence of O₂.
As shown in Scheme 23, the alternative route B is able to produce a N—N bond through double single-electron transfer [68].

![Scheme 22](image)

**Scheme 22.** A synthetic route for the preparation of 1H-indazole derivatives 80 via C—H amidation and N—N bond formation.

![Scheme 23](image)

**Scheme 23.** A rational mechanism for the synthesis of 1H-indazole derivatives 80 via C—H amidation and N—N bond formation.

An efficient strategy for the synthesis of 3-acyl-(2H)-indazoles 90 via an Rh(III)/Cu(II)-catalyzed [4 + 1] annulation of azobenzenes 10 with α-carbonyl sulfoxonium ylides 89 has been developed by Cheng et al. (Scheme 24). The sequential catalytic aromatic C—H bond activation and cyclization steps in which sulfoxonium ylides 89 served as efficient and stable carbene precursors are highly important in the chemical synthesis of the corresponding indazoles 90. The model reaction produced 3-acyl-(2H)-indazole in a 20% yield in the absence of Cu(OAc)₂ in DCE under air at 100 °C within 24 h. The scope and limitation of azobenzene derivatives 10 and α-carbonyl sulfoxonium ylide derivatives 89 as starting materials were explored, as illustrated in Scheme 24. Some functional groups on azobenzene rings, such as methyl, methoxy, Cl, F, i-Pr, and Br, were very compatible under the optimized reaction conditions with moderate to high yields. The reaction progressed efficiently
by using sterically hindering substituents on the azobenzene ring under standard reaction conditions. As expected, with m-substituted substrates, the corresponding indazole derivatives were produced at the less hindered position. Unfortunately, the 4-nitro analogue of azobenzene failed to work. Unsymmetrical azobenzenes with Me and MeO groups were also explored, indicating the orientation of electron-rich aryl rings for the C–H activation and cyclization transformations. Next, sulfoxonium ylides with both electron-donating and electron-withdrawing substitutions were investigated. Several substitutions such as Me, MeO, Cl, NO₂, and t-Bu as well as alkyl- and heteroaryl-substituted analogues were found to be appropriate under standard reaction conditions. The reaction displayed a broad substrate scope, moderate to excellent yields, and tolerance to various substituents [69].

Initially, the catalyst is activated in the presence of AgSbF₆. Next, the Rh(III) catalyst activates the C–H bond of azobenzenes 10 to deliver five-membered cyclorhodium species 91. Rh(III) intermediates 92 are generated by the insertion of sulfoxonium ylides 89 to the rhodium species. Six-membered rhodacycle species 94 are obtained via the α-elimination of DMSO followed by the migratory insertion of carbene complex 93. The sequence of keto-enol tautomerization, the insertion of a C=C bond into the Rh-N bond, and the oxidation/ aromatization in the presence of a copper catalyst produces 95, 96, and final products 90, respectively (Scheme 25) [69].

The Rh(III)-catalyzed ortho-C–H bond activation of azobenzenes 10, followed by the intramolecular annulation transformation of azobenzenes 10 between sulfoxonium ylides 89, has been reported for the synthesis of 2H-indazole derivatives 90 in good to excellent yields with broad substrate scope, good selectivity, and good functional-group tolerance (Scheme 26). The combination of [RhCp*Cl]₂, AgSbF₆, Cu(OAc)₂, CuCO₃·Cu(OH)₂, and DCE at 110 °C for 24 h under air atmosphere in reaction tubes was found to be effective for this transformation. Next, the scope of these annulation reactions was examined, and the results are summarized in Scheme 26. The reactions using electron-rich para-substituted azobenzenes (such as Me, Et, and t-Bu) afforded the expected indazoles in excellent yields. Furthermore, the electron-rich meta- and ortho-substituted azobenzenes (such as OMe and Br at the meta-position or OMe, Me, and Et at the ortho-position) were explored, and a variety of indazoles were successfully produced in good to excellent yields. An electron-deficient azobenzene (CO₂Et at the para-position) was also applicable to this.
reaction, which afforded the corresponding product in a moderate yield, while electron-deficient meta-substituted azobenzenes (Br and F in the meta-position) were proven to be relatively less efficient for this transformation. Although a sterically congested disubstituted azobenzene (two methyl groups at the 2- and 5-positions) did not give the desired product, by employing disubstituted azobenzenes (two methyl groups at the 2- and 4-positions or two methyl groups at the 2- and 3-positions) as substrates, the corresponding products were achieved in good yields. The steric congestion of product 90f may be the reason of why 2,5-disubstituted azobenzene did not work in the reaction [70].

Scheme 25. A plausible mechanism for the synthesis of 3-acyl-(2H)-indazoles 90 via the [4 + 1] annulation of azobenzenes 10 with a-carbonyl sulfoxonium ylides 89.

Scheme 26. A synthetic route for the preparation of 2H-indazoles 90 through the annulation reaction of azobenzenes 10 with sulfoxonium ylides 89.
A plausible reaction mechanism for the Rh(III)/Cu(II) catalyzed synthesis of indazoles 90 is shown in Scheme 27. The mechanism comprises the coordination of the azo group to the Rh(III) catalyst, C–H bond activation, the coordination of sulfoxonium ylides 89, and the α-elimination of DMSO to give a reactive α-oxo Rh-carbene species 93. In the next step, migratory insertion and, subsequently, protonation occur to provide 97. Base-mediated intramolecular annulation and oxidation by copper or O₂ take place to afford 3-acyl (2H)-indazoles 90 [70].

Scheme 27. A possible mechanism for the synthesis of 2H-indazoles 90 through the annihilation reaction of azobenzenes 10 with sulfoxonium ylides 89.

The successful use of N-nitrosoanilines 100 in which the N-nitroso group plays a dual role as a traceless directing group and an internal nitrosation reagent in the reaction with several sulfoxonium ylides 89 through the Rh(III)/Cu(II)-catalyzed C–H activation and cyclization reaction was demonstrated by Huang et al. in 2020 (Scheme 28). Thus, the subsequent oxidative [4 + 1] annulation reactions of various N-nitrosoanilines 100 with sulfoxonium ylides 89 were conducted using Cp*Rh(OAc)₂ · H₂O (10 mol%), Cu(OAc)₂ (0.5 eq), and Ag₂O (1 eq) in TFE at 100 °C for 10 h. This simple and efficient strategy could be utilized for the synthesis of diverse substituted indazole N-oxides 101 with powerful reactivity, good functional-group tolerance, moderate to good yields, and atom- and step-economic reactions under mild conditions. Notably, this unique annihilation approach indicates a previously unobserved reactivity model for the N-nitroso functional group. The scope and limitations of this cascade acylmethylation/annulation transformation were tested using the reaction of a variety of N-nitrosoanilines 100 and sulfoxonium ylides 89. In general, a diverse array of functional groups (various electron-donating groups and halides) on the o-, m-, and p- positions of N-nitrosoaniline rings were possible under these conditions. In the case of m-Cl-substituted N-nitrosoanilines, a mixture of regioisomers was obtained. N-nitrosoanilines bearing two substituents exhibited moderate performances, providing desired products 101a and 101b. N-alkyl-N-nitrosanaphthylamines with different alkyls substituted on the amino group and substituted N-nitroso-tetrahydroquinoline were also tolerated, furnishing good to high yields of products 101c–e. It is noteworthy that sulfoxonium ylides with one or two substitutions on different positions of the aryl ring were very good partners under the standard conditions, achieving moderate to high yields.
The naphthyl-derived substrates heterocyclic and alkyl ylides were next examined using this catalytic system, showing moderate yields of 101i-j [71].

Scheme 28. A synthetic route for the preparation of substituted indazole N-oxides 101 via the tandem acylmethylation/annulation of N-nitrosoanilines 100 with sulfoxonium ylides 89.

A reasonable mechanism is illustrated in Scheme 29. In the first step, coordination/C−H activation with a Rh catalyst affords complex 102. The insertion of sulfoxonium ylide to the metal center of the previous complex produces a Rh(III) intermediate 103, followed by releasing a DMSO molecule from 103. Subsequently, the resulting rhodium α-oxo carbene complex 104 carry out the migratory insertion of the Rh-C bond to provide a six-membered rhodacyclic intermediate 105. In the next step, the acylmethylated intermediate 106 is obtained by the protonolysis of the Rh-N and Rh-C bonds. Subsequently, the final product 101 is generated from 106 by Cu(OAc)$_2$-mediated intramolecular annulation and oxidation in the presence of either Ag$_2$O or Cu(II) additives [71].

Scheme 29. A rational mechanism for the synthesis of substituted indazole N-oxides 101 via the tandem acylmethylation/annulation of N-nitrosoanilines 100 with sulfoxonium ylides 89.
Interesting molecular architectures were achieved by the Rh(III)/Cu(II)-catalyzed [4 + 1] cyclization of azobenzenes 10 with α-Cl ketones 109 in excellent yields for more than 30 examples (Scheme 30). The generality and scope of 3-acyl-2H-indazoles synthesis by performing the annulation of azobenzenes 10 with α-Cl ketones 109 were screened. It was shown that this transformation could tolerate diverse starting materials bearing a variety of functional groups in both azobenzenes 10 and α-Cl ketones 109 and provided the desired products 90 in good to excellent yields. Generally, azobenzenes 10 with both electron-donating and -withdrawing functional groups at the α, m, and p positions as well as alkyl-, alkoxy-, or halogen-substituted azobenzenes afforded the desired products in moderate to good yields. It was observed that m-F substitution led to a mixture of two regioisomers, 90aa and 90aa’, in which the less hindered position was favored. Meanwhile, an unsymmetrical azobenzene-containing methoxy group was annulated with α-Cl ketone containing hydroxy at the m position to form an anti-inflammatory agent 90bb with a high yield in a regioselective manner. Alkyl or heterocyclic azo compounds as substrates have not been applicable in this catalytic system. Various monosubstituted α-Cl ketones and disubstituted α-Cl ketones were able to undergo this cascade C–H activation/annulation reaction smoothly with azobenzenes to afford the corresponding products in moderate to excellent yields. As a result, substituent effects have no large effect on the yields, although strong electron-withdrawing groups on the α-Cl acetophenone aryl ring seem to be more practical for this cascade reaction. Moreover, alkenyl- and naphthyl-substituted scaffolds 90cc–dd were also successfully used in the reaction to afford the corresponding products in moderate to good yields.

Reactions occurred with high efficiency and with extensive functional-group tolerance under mild reaction conditions. It is worth noting that the efficient production of 3-acyl-2H-indazole 90bb with anti-inflammatory activity in one step ensures the practicability of this approach [72].

A reasonable mechanistic pathway is shown in Scheme 31. The subsequent coordination of azobenzenes 10 with the Rh(III) catalyst and C–H bond activation give a five-membered rhodacyclic intermediate 91. The α-Cl acetophenones 109 coordinate to Rh(III), followed by C–C bond formation. N-protonation takes place to produce α-arylketone species 97 along with the regeneration of the Rh(III) catalyst. Enol-tautomerization,
intramolecular annulation, and oxidation transformations occur in the presence of Cu(II) species to produce target molecules 90 [72].

Scheme 31. A rational mechanism for the synthesis of 3-acyl-2-H-indazoles 90 via the annulation of azobenzenes 10 with α-Cl ketones 109.

An efficient and novel C–H activation and C–H/C–H cross-coupling strategy has been developed to realize the synthesis of functionalized 1H-indazoles 113 from easily accessible aldehyde phenylhydrazones 112 (Scheme 32). Optimization studies showed that the best yields could be achieved by performing the reaction using a catalytic amount of (RhCp*Cl)2/AgOTf, with Cu(OAc)2 as the oxidant and K2CO3 as the base at 120 °C in 1,2-dichloroethane. Initially, the scope of the substrates with respect to different N-alkyl (Me, Et, and Bn) or aryl groups was tested. It was observed that these functional groups had only a moderate effect on the reaction, generating the desired compounds in very good to high yields. Whereas benzaldehyde derivatives 112 containing different electron-rich and electron-poor functional groups on the para- or meta-positions of the aryl ring yielded good to high yields of the corresponding indazoles, ortho-substituted benzaldehyde derivatives displayed less efficacy in this reaction, delivering the corresponding indazole 113 in a low yield. Heteroaromatics such as furan- and thiophene-aldehyde starting materials were also well-tolerated under this catalytic strategy, leading to the expected indazoles in acceptable yields. In the case of substrates substituted at the meta-position of the N-aryl ring, the regioselectivity was moderate, showing a mixture of products 113a and 113a' that had reacted at both positions. Some complex products 113b–k synthesized by this methodology are illustrated in Scheme 32. In order to highlight the importance of this strategy, the authors extended the procedure to an easy preparation of certain important and bioactive products. These significant products displayed good 5-HT4/5-HT3 receptor antagonist activity. This transition metal strategy reveals good scalability, good functional-group compatibility, wide substrate scope, and moderate to high yields under mild reaction conditions. The combination of mechanistic experiments and DFT calculations indicates C(aryl)-H bond activation followed by a C(aldehyde)-H bond activation, and the reductive elimination process occurs to form a variety of targeted products 113 [73].
2.10. Synthesis of Indazoles Using Rhodium, Silver, Copper, and Zinc Salts

Efficient rhodium(III)-catalyzed regioselective C–H activation/cyclization of an azoxy compound 31 with an alkyne 114 as the coupling partner has been realized via a [4 + 1] cycloaddition rather than a normal [4 + 2] mode (Scheme 33). The processes of cyclative capture, oxygen-atom transfer, and C≡C triple bond cleavage are the key steps of this reaction. The coupling of azoxy compounds 31 with alkenes 114 was conducted using [Cp*RhCl₂]₂ (2.5 mol%) as a precatalyst, AgSbF₆ (10 mol%), Cu(OAc)₂ (1.0 equiv), Zn(OTf)₂ (20 mol%), and HFIP (1.0 mL) under N₂ at 80 °C for 12 h. It seems AgSbF₆ and Cu(OAc)₂ are exploited to activate the Ru catalyst before or during the catalytic cycle. The authors mentioned no exact role for the Lewis acid Zn(OTf)₂. Initially, the generality of this catalytic system was studied with respect to the azoxy substrates 31. The reactions with symmetrical azoxybenzenes containing electron-rich substituents including Me, t-Bu, and OMe as well as electron-withdrawing substituents such as F, Cl, Br, and COOEt proceeded smoothly to give indazoles 115 in moderate to excellent yields. It is worth noting that a variety of meta-substituted azoxybenzenes were used in the reaction to produce the corresponding products 115 in good to high yields while reacting at the less sterically hindered position. The strategy could also be extended to unsymmetrically substituted diaryldiazene oxide substrates 31 (leading to 115a–n). Even the monoaryl-diazene oxide substrates were well-tolerated and successfully reacted with diphenylacetylene to produce the expected indazoles 115g–k in satisfactory yields. While the reaction was compatible with a wide range of azoxybenzenes, substrates functionalized by substituents such as Me, OMe, Cl, and Br at the ortho-position of the NO-phenyl ring failed to give corresponding products, probably due to steric congestion. Furthermore, the scope and limitations of the reaction were tested by the exploitation of alkyne substrates. As summarized in Scheme 33, different symmetrical diarylalkynes with both electron-rich and electron-poor substituents were successfully
utilized under the optimized reaction conditions, which gave indazoles 115 in moderate to good yields. It is noteworthy that this catalytic system showed complete compatibility with diarylacetylene substrates bearing 2-naphthyl, 2-thienyl, or 3-pyridyl scaffolds as well as dialkylalkyne substrates. Bis(trimethylsilyl)acetylene or monosubstituted alkyne substrates failed in this [4 + 1] annihilation reaction. This strategy demonstrates various outstanding features, such as broad substrate scope, good functional-group tolerance, and operational convenience, which enable regioselective access to different 2,3-disubstituted 2H-indazoles 115 in moderate to high yields [74].

A plausible reaction mechanism for the Co(III)/Cu(II)-catalyzed synthesis of 1H-indazole derivatives 115 via the [4 + 1] annulation of azoxy compounds 31 with alkynes 114.

![Scheme 33](image)

**Scheme 33.** A synthetic route for the preparation of 2H-indazole derivatives 115 via the [4 + 1] annihilation of azoxy compounds 31 with alkynes 114.

### 2.11. Synthesis of Indazoles Using Cobalt and Copper Salts

A Co/Cu-catalyzed C–H activation/oxidative coupling of imidate esters 1 with anthranils 116 in DCE at 100 °C (sealed tube) under N₂ atmosphere within 20 h for the synthesis of 1H-indazole derivatives 117 was described by Li et al. (Scheme 34). In this paper, anthranils 116 were exploited as novel bifunctional aminating reagents and organic oxidants under Co/Cu catalysis, affording a broad range of 1H-indazoles 117 in high yields (up to 99% yield) with excellent selectivity. It seems that the formation of the N–N bond involves Cu catalysis. Initially, anthranil 116a was reacted with a range of imidates containing both electron-donating and withdrawing functional groups at the para-site of the phenyl ring. The results summarized in Scheme 34 demonstrate that this methodology is useful for the coupling of 116a with a range of functionalized imidates 1. The meta-substituted imidates were examined for the synthesis of the product 117 under standard conditions, with the C–N/N–N coupling of imidates 1 with anthranil 116a taking place at the less hindered position. In the next step, the substrate scope of anthranil derivatives 116 under standard reaction conditions was investigated. Various substituents, including halogens and phenyl on different positions of the anthranil ring and a phenyl group substituted into the 3-position of the anthranil ring, were all applicable, leading to the desired indazole derivatives in satisfactory yields. Although nonsubstituted anthranil was an effective substrate, the reaction rate was decreased. In this case, the utilization of pivalic acid additive facilitated the C–H activation transformation. When the reaction was carried out with 2-azidobenzaldehyde precursors instead of anthranils, poor results were obtained [75].

A plausible reaction mechanism for the Co(III)/Cu(II)-catalyzed synthesis of 1H-indazoles 117 from the coupling of imidates 1 with anthranils 116 is shown in Scheme 35. The mechanism comprises the cyclometalation of the imidate 1, the coordination of anthranil to intermediate 118, intramolecular N–O bond cleavage forming 120, the migratory
insertion of the Co–aryl bond of 122 into the nitrone, and the coordination of an imidate to 121, along with releasing the aminated intermediate 122. In a successive catalytic cycle, the coordination of Cu(OAc)$_2$ to 122 and a double single-electron transfer occur to form target indazole 117 [75].

Scheme 34. A synthetic route for the preparation of 1H-indazoles 117 via the C–N/N–N coupling of imidates 1 with anthranils 116.

Scheme 35. A rational mechanism for the synthesis of 1H-indazoles 117 via the C–N/N–N coupling of imidates 1 with anthranils 116.
2.12. Synthesis of Indazoles Using Palladium Salts

The Pd-catalyzed C−H activation/annulation of substrates 127 was achieved by Charette in 2015. A series of 3-aminindazoles 128 was produced with moderate to high yields (Scheme 36). In this reaction, the intermediate amidrazenes were initially synthesized using an easy route from amide substrates 129 via (1) triflic anhydride activation and (2) condensation with hydrazines, to then (3) undergo a Pd-catalyzed C−H amination reaction. Structurally diverse amidrazenes bearing various functional groups on the arene or nitrogen substituents revealed acceptable reactivity, yielding the expected indazoles 128. As illustrated in Scheme 36, starting substrates with several cyclic and acyclic amine scaffolds can be applicable in the reaction. Various types of para-substituted substrates were investigated, delivering the desired indazoles 128 in yields of 15%−70%. In the case of a meta-substituted substrate (127 substituted with OMe at the meta-position), a 4:1 mixture of regioisomers was obtained. Unfortunately, ortho-substituted substrates did not work in the reaction. Notably, the reaction was also effective using a substrate containing a 2-benzothienyl scaffold, achieving 3-aminindazole fused heterocycle 128c. The author offered no mechanistic explanation, but this may likely involve classical Pd-catalyzed C−H activation followed by cyclization. The current strategy offers an efficient procedure for the synthesis of 3-aminindazoles 128, which are easily synthesized from amide substrates in two steps [24].

![Scheme 36. A synthetic route for the preparation of 3-aminindazoles 128 from tertiary amides 127.](image)

2.13. Synthesis of Indazoles Using Palladium Salt in the Presence of Cobalt Salts

An efficient Pd-catalyzed C−H functionalization/nitration/cyclization sequence for the synthesis of 3-nitro-1-(phenylsulfonyl)-1H-indazole derivatives 130 under mild reaction conditions was reported for the first time by Xia et al. (Scheme 37). The chelate-assisted cleavage of two C−H bonds is the key step in this transformation. The reaction was conducted using substrates 129 in the presence of 20 mol% Pd(OAc)2 and 1.5 eq Co(NO3)2·6H2O in DCE in a sealed tube at 110 °C for 4 h. Co(NO3)2·6H2O has been used as an efficient nitration agent. A variety of functional groups on the para- and meta-positions of the benzylidene part were well-tolerated with little electronic dependence, providing, in most cases, 3-nitro-1-(phenylsulfonyl)-1H-indazoles 130 in excellent yields with high selectivity. The substrates substituted at the ortho-positions of the benzylidene scaffold underwent the C−H nitration and intramolecular C−H activation sequence uneventfully to give the target indazoles in more than 80% yields. As shown in Scheme 37, disubstituted substrates could also be successfully employed to deliver the indazoles 130a−e. However, the di-MeO substrate failed to provide the desired product under the optimized reaction conditions. It was also found that various substrates containing substitutions at different positions of the phenylsulfonyl group were all tolerated to provide the desired 3-nitro-1-(phenylsulfonyl)-1H-indazole derivatives 130 with
high yields for most cases. In the case of 4-NO$_2$, the reaction did not work under optimized reaction conditions (only a trace amount of target product 130f was produced). In general, this transformation proceeded smoothly and could tolerate various substituents, providing easy access to a series of new indazoles [76].

Scheme 37. A synthetic route for the preparation of 3-nitro-1-(phenylsulfonyl)-1H-indazole derivatives 130 via the direct C–H nitration and intramolecular C–H functionalization of 129.

A plausible mechanism for the synthesis of 130 is outlined in Scheme 38. The mechanism initially comprises the complexation of Pd(OAc)$_2$ with the anionic intermediate 131, the combination of in situ-generated NO$_2$• with intermediate 132, and oxidation and reductive elimination to form intermediate 135 and Pd(OAc)$_2$. Next, the coordination of intermediate 135 with Pd(OAc)$_2$ followed by C–H bond activation affords six-membered Pd(II) species 137. Finally, oxidation to a Pd(IV) species 138 by air and reductive elimination take place to produce corresponding product 130 and the Pd(OAc)$_2$ [76].

Scheme 38. A rational mechanism for the synthesis of 3-nitro-1-(phenylsulfonyl)-1H-indazole derivatives 130 via the direct C–H nitration and intramolecular C–H functionalization of 129.
2.14. Synthesis of Indazoles Using Palladium, Cerium, and Iron Salts

A novel, efficient, rapid, and versatile protocol to generate pyrido[1,2-b]indazole derivatives 140 through the C–H activation/azidation of arylpyridines 139 has been developed by Jiao et al. (Scheme 39). The synthetic approach involves C–H azidation and N-N bond formation, which was carried out at 100 °C using 15 mol% Pd(OAc)$_2$, 20 mol% FeCl$_2$, and Ce(SO$_4$)$_2$ in DMSO at 100 °C under O$_2$ atmosphere for 79–82 h. The scope of this reaction was first studied using the 3-methoxy-2-arylpypyridine substrate and its derivatives. The substituent effect (electron-rich or electron-poor substituents) on the 3-methoxy-2-arylpypyridine substrate was checked, showing a moderate effect on the yields. The 3-methyl-2-phenylpyridine substrate was explored as a further substrate, affording a satisfactory yield of the desired product. 4-Methyl, 5-methyl, or 3,5-dimethyl-substituted-2-phenylpyridines could also be successfully employed to provide the desired pyrido[1,2-b]indazoles in acceptable yields. It is worth noting that sterically hindered substrates could also be utilized, only providing the corresponding pyrido[1,2-b]indazoles in moderate yields. As expected, substrates with either electron-rich substituents such as Me and t-Bu or electron-poor substituents such as F and Cl at the para-position of the phenyl ring also proved to be suitable for this Pd-catalyzed tandem C–H azidation and N-N bond formation. Finally, 1-phenylisoquinoline yielded the corresponding product 140g under optimized reaction conditions in a good yield. Involving direct C–N (via 2-pyridyl-directed Pd-catalyzed azidation) and N–N (via concerted nitrogen loss/ring closure) formations, this procedure proceeds under mild conditions, thus demonstrating a methodology for the synthesis of [b]fused indazoles in a single step [77].

2.15. Synthesis of Indazoles Using Palladium, Copper, and Silver Salts

An efficient procedure for the mild synthesis of 3-aryl/alkylindazoles 142 catalyzed by palladium via a direct C–H activation/intramolecular amination sequence of hydrazide substrates 141 was presented by Hiroya (Scheme 40). This catalytic C–H activation/intramolecular amination reaction was performed involving 10 mol% Pd(OAc)$_2$, 1 equiv Cu(OAc)$_2$, and 2 equiv AgOCOCF$_3$ in DMSO at 50 °C for 10–24 h using hydrazide substrates 141. This reaction was investigated using different substrates containing both electron-donating and -withdrawing functional groups on the benzene rings. As illustrated in Scheme 40, the transformation with two para-methoxy substituents gave only a 13% yield of product 142a. In contrast, the compound containing two para-methyl substituents provided the expected product 142b with a 73% yield. Surprisingly, the compound containing two meta-methoxy substituents proceeded successfully to afford only one product in a regioselective manner. Next, the starting materials containing two different groups at the meta-position reacted only at the 6-position on the more electron-donating aromatic ring, evidencing that both steric and electronic factors affect the outcome (see the products 142d–f). The products 142 were achieved in acceptable yields using diverse monosubstituted substrates. As expected,
the reaction pathway was completely dependent on the nature of the functional groups on the benzene rings. The electron-rich donating substituents mainly oriented the reaction towards the A1 arene 142g-h, whereas hydrazones containing an electron-poor substituent on the arene produced the corresponding indazoles on the nonsubstituted A2 arene 142d-f. Notably, a C−H activation/intramolecular amination sequence took place preferentially on the A2 arene when para-substituted hydrazones containing electron-rich or -poor substituents were explored 142m-n [78].

Scheme 40. A synthetic route for the preparation of 3-aryl/alkylindazoles 142 via the C−H activation/intramolecular amination reaction.

2.16. Synthesis of Indazoles Using Cobalt Salts

A novel efficient synthesis of indazoles 71 via sequential C−H bond functionalization/addition/cyclization was reported by Ellman (Scheme 41). The target products 71 were synthesized via a convergent one-step route using a novel air-stable cationic Co(III) catalyst. The reaction has been previously performed using 5 mol% (Cp*RhCl2)2 and 20 mol% AgSbF6 in dioxane at 80 °C for 24 h. Although this system is efficient for the synthesis of indazoles, this chemistry displays some disadvantages, especially using an expensive Rh catalyst as well as the availability of Rh in the nature [66]. The reaction was carried out in the presence of 10 mol% co-catalyst 143 and 10 mol% AcOH in 1,4-dioxane (2.0 M) at 100 °C for 24 h and afforded the desired products 71 mostly with acceptable yields. First, the substrate scope of azobenzenes 10 was explored. Here, several symmetric and unsymmetrical azobenzenes containing electron-rich and electron-poor substituents were all well-tolerated and delivered the desired compounds in a regioselective manner. However, in the case of the para-nitro or meta-methoxy substituted azobenzenes, a mixture of products 71aa, 71aa’, 71ab, and 71ab’ was obtained. Different 3,5-disubstituted azobenzene substrates bearing a variety of substituents with donating or deficient natures were proven to afford satisfactory yields involving the less sterically hindered ring in the cyclization. Moreover, aromatic aldehydes containing electron-rich and electron-poor substituents as well as heteroaromatic aldehydes successfully reacted with azobenzene to form the corresponding indazoles in acceptable yields. The results of this study are illustrated in Scheme 41 [79].
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Scheme 41. A synthetic route for the preparation of indazoles 71 via C–H bond functionalization/addition/cyclization cascades.

2.17. Synthesis of Indazoles Using Rhenium Salt in the Presence of Sodium Acetate

Wang and co-workers employed the first Re catalysis for the [4 + 1] annulation of azobenzenes 10 with aldehydes 70 to furnish 2H-indazoles 71 using catalytic amounts of Re$_2$(CO)$_{10}$ and NaOAc in toluene at 150 °C within 50–72 h in an oven-dried Schlenk tube (Scheme 42). In comparison to similar works performed by Rh [66] and Co [79] catalysts, Re is less expensive than Rh but more expensive than a Co catalyst. It should be noted that Co and Rh catalysts could provide milder conditions for the reaction. The acetate could potentially enable an acceleration effect in the Re-catalyzed C–H activation of azobenzenes 10. Initially, the scope of aldehyde substrates 70 was tested with both electron-rich and -poor functional groups on the aromatic ring, producing the target products in moderate to high yields. A wide variety of aldehydes 70 with substituents such as F, Cl, Br, MeS, CO$_2$Me, and CF$_3$ also coupled with different azobenzenes 10 to give corresponding products 71 in satisfactory yields. The reaction progressed smoothly using naphthaldehydes and thiophene-2-carbaldehyde to afford indazole derivatives 71 in good yields. When aliphatic aldehydes were introduced into this reaction, the yield was decreased. The scope and limitations of the reaction were screened by focusing on azobenzene derivatives 10. The reaction proceeded smoothly with a variety of para-substituted azobenzenes, giving the corresponding indazoles 71 in good to high yields. Moreover, a variety of substituents on the ortho-position of azobenzenes were well-tolerated under standard conditions. In the case of disubstituted azobenzene (two methyl groups at the 2- and 4-positions of both rings), the reaction efficiently underwent the C–H functionalization/[4 + 1] annulation and generated the expected product in a high yield. Meanwhile, meta-methyl azobenzene produced a moderate yield of the corresponding product with high regioselectivity. The structures of some target products are shown in Scheme 42. No significant effect on the yield was observed when azobenzene was replaced with a bulky (mesityl) analog [80].
Scheme 42. A synthetic route for the preparation of indazole derivatives 71 via the [4 + 1] annulation of azobenzenes 10 with aldehydes 70.

A reasonable reaction mechanism for the rhenium-catalyzed [4 + 1] annulation of azobenzenes 10 with aldehydes 70 is shown in Scheme 43. First, C−H is activated in the presence of Re$_2$(CO)$_{10}$/NaOAc to afford a cyclic Re(I)-complex 144. In the next step, complex 145 bearing aldehyde is generated via an exchange between a CO ligand and an aldehyde. Finally, an irreversible aldehyde insertion to the Re−aryl bond, protonation, and an intramolecular nucleophilic substitution transformation followed by rearomatization produce the final product 71 [80].

Scheme 43. A rational mechanism for the synthesis of indazole derivatives 71 via the [4 + 1] annulation of azobenzenes 10 with aldehydes 70.
2.18. Synthesis of Indazoles Using Copper Salts

A highly regioselective synthesis of indazoles 151 was reported by Glorius via a copper-catalyzed C–H amidation process transformation. In this reaction, azides 150 were used as amino sources, and no oxidizing agent was used. The reaction exploited amidines or imine 149 to perform a tandem C–N/N–N bond-forming transformation for the production of indazole derivatives 151 in one pot (Scheme 44). All annulation reactions were carried out in 1,2-dichlorobenzene (DCB) at 115 °C using copper(I) thiophene-2-carboxylate (CuTc) (30 mol%) and 1,1,1,3,3,3-hexafluoro-2-propanol (HFIP) (1.0 eq) under Ar in a sealed tube. The N-tert-butylarylamidine substrates 149 bearing Me, MeO, Cl, or Br have been successfully employed in the Cu-catalyzed synthesis of N-tert-butyl-3-aminooindazoles using coupling with TsN₂ via a tandem C–N/N–N bond-forming reaction. A more detailed explanation of the mechanism includes a Cu-metallacycle, then nitrene coordination, migratory insertion, and reductive amination. The reaction of both tosyl-protected and free benzimidamide failed to form the corresponding products under optimized conditions. In addition, diphenylmethanimine as the starting material was examined, showing the expected indazole formation [81].

![Scheme 44](image)

Scheme 44. A synthetic route for the preparation of indazole derivatives 151 via C(sp²)−H amidation 149 with azides 150 as amino sources.

An atom-economical protocol for the practical preparation of indazoles 153 through the Cu-catalyzed direct aerobic oxidative C(sp²)−H amination of benzophenone hydrazones 152 has been illustrated by Jiang et al. (Scheme 45). The synthesis of indazoles 153 was carried out successfully through an aerobic C(sp²)−H activation for C−N bond formation in the presence of catalytic amounts of Cu(OAc)₂ and DABCO in DMSO within 12 h in 70–86% yields. The substrate scope of the reaction was then tested. As shown in Scheme 45, the reactions of substrates 152 with substituents such as F, Cl, CF₃, and CN were then explored. These cascade reactions proceeded smoothly and afforded the corresponding indazoles 153 in moderate to good yields under standard conditions. Notably, substrates 152 containing an electron-rich substituent produced the desired indazoles in relatively higher yields than those with electron-poor substituents. It is worth noting that by using benzophenone hydrazone 152 containing two electronically unsymmetrical aromatic groups, a mixture of isomers was obtained, indicating that the reaction does not proceed in a regioselective manner. Some complex products synthesized by this Cu-catalyzed direct aerobic oxidative C(sp²)−H amination are illustrated in Scheme 45. The procedure has the privilege of good to high yields and tolerates a variety of substituents.

First, the Cu(OAc)₂ is probably coordinated with substrate 152, resulting in a Cu−N adduct 154. Subsequently, the C−H activation and C-N bond-forming transformation take place to afford the target molecules 153 (Scheme 46) [23].
Scheme 45. A synthetic route for the preparation of indazole derivatives 153 via aerobic C(sp²)−H functionalization/C−N bond formation. 

First, the Cu(OAc)₂ is probably coordinated with substrate 152, resulting in a Cu−N adduct 15. Subsequently, the C−H activation and C−N bond forming transformation take place to afford the target molecules 153 (Scheme 46) [23].

Scheme 46. A rational mechanism for the synthesis of indazole derivatives 153 via aerobic C(sp²)−H functionalization/C−N bond formation.

3. Conclusions

The N-heterocycles show a wide structural variety that is beneficial for the investigation of further therapeutic agents for enhancing the pharmacokinetics and other physicochemical properties. Indazole derivatives are a significant category of N-heterocyclic five-membered systems that display a broad range of applications in biology, chemistry, and materials science. The investigation of indazole derivatives has become a rapidly evolving and increasingly active subject in therapeutic science. Meanwhile, transition-metal-catalyzed C–H activation/annulation strategies for the one-pot synthesis of indazoles has been growing steadily in synthetic organic chemistry. They are a superior class of trans-
formations that serve as powerful tools to access a wide range of indazoles. Numerous outstanding results have been reported for the synthesis of these structurally diverse indazole frames via transition-metal-catalyzed C–H activation/annulation reactions. The results of these articles have not previously been thoroughly studied or reviewed. This review article presents several atom-economical strategies for synthesizing indazole derivatives from accessible substrates through metal-catalyzed C–H functionalization and annulation transformations. The mentioned reactions feature easily available substrates, relatively good to high yields, large functional-group tolerance, diversity, and a high complexity of products under mild reaction conditions, furnishing practically useful methods to access structurally diverse indazoles.

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