Short Communication

Heavy metal utilization by locally available plants

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Abstract
The present paper deals with removal of heavy metal namely Copper from polluted water using plant species oscimum and mint. The experiment was divided into 2 groups. Group one as control and group II treated as experimental which received 4ppm of Copper sulphate. In the experimental plants were exposed to 4ppm Cu for 24 and 48 hrs. And the amount of Cu depleted from water was analyzed. From the experiment it is evident that the Cu from water was utilized by the plants was more during 48 hrs than in 24 hr.

Keywords: Metal, Plant, Copper

1. Introduction
Smoking Biological approaches of remediation include use of microorganisms to detoxify the metals by valence transformation, extracellular chemical precipitation use of special type of plants to decontaminate soil or water by inactivating metals in the rhizosphere or translocating them in the aerial parts. This approach is called phytoremediation, which is considered as a new and highly promising technology for the reclamation of polluted sites and cheaper than physicochemical approaches[3][4][6]. Phytoremediation, also referred as botanical bioremediation[2], involves the use of green plants to decontaminate soil, water and air. It is an emerging technology that can be applied to both organic and inorganic pollutants present in the soil, water or air[7].

Phytoremediation is a technology that uses plants for cleanup of contaminated environment. The concept of using plants to cleanup of contaminated environment is not new practice. About 300 year ago plants where used for treatment of waste water. These plants accumulate high levels of pollutants and metals. For example plants like mint and, oscimum are commonly used for phytoremediation of metals like zinc, copper, manganese etc.

2. Materials and Method
Estimation of Copper was carried by Standard Methods[1]. Waste water and water properties. Oscimum and Mentha saplings of same age are collected and exposed to 4 ppm (group II) Copper sulphate solution for 24 and 48 hours. A control is also maintained simultaneously. After stipulated period of exposure the water from group I and II are analysed for Copper levels for which 50 ml of water sample is collected and to it 2 drops of Euriochromic indicator and 10 ml of Ammonium buffer is added. The colour of the sample turns vine red colour then sample is titrated against EDTA. If the sample turns to blue it indicates presence of Copper if not copper is considered to be absorbed or used by plant.

3. Results and Discussion
From the results it is evident that after 24 and 48 hours of exposure, the levels of Cu in water sample was depleted more in 48 hrs exposure than in 24 hrs. Mint plants absorbed more amount of Cu than Oscimum (Table 1).

Rhizofiltration is the removal of pollutants from the contaminated waters by accumulation into plant biomass. Several aquatic species have been
identified and tested for the phytoremediation of heavy metals from the polluted water. The roots of Indian mustard are found to be effective in the removal of Cd, Cr, Cu, Ni, Pb and Zn, and sunflower can remove Pb, U, Cs-137 and Sr-90 from hydroponic solutions[5][8][9].

This study aims to verify the possibility of using fast growing plants to decontaminate water with high Cu contamination levels and to test its hyperaccumulating performances. Decrease in the levels of Cu in water sample indicates plant accumulated the metal. To date the available methods for the recovery of heavy metals from plant biomass of hyperaccumulators are still limited. Traditional disposal approaches such as burning and ashing are not applicable to volatile metals; therefore, investigations are needed to develop new methods for effective recovery of metals from the hyperaccumulator plant biomass.

Table 1: The amount of Copper absorbed by plant

| S. No. | Name of the plant  | 24hrs     | 48hrs     | control |
|--------|--------------------|-----------|-----------|---------|
| 1      | Mintha piperita    | 1.75±0.4  | 2.15±0.7  | 1.67±0.5 | 4ppm    |
| 2      | Oscimum species    | 1.08±0.2  | 1.67±0.5  | 4ppm    |

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