Electrochemical carbon fiber-based technique for simultaneous recordings of brain tissue PO$_2$, pH, and extracellular field potentials

Patrick S. Hosford$^{a, b, *}$, Jack A. Wells$^c$, Isabel N. Christie$^a$, Mark F. Lythgoe$^c$, Julian Millar$^d$, Alexander V. Gourine$^{b, **}$

$^a$ Centre for Cardiovascular and Metabolic Neuroscience, Department of Neuroscience, Physiology & Pharmacology, University College London, London, UK
$^b$ William Harvey Research Institute, Barts and the London School of Medicine and Dentistry, Queen Mary University of London, London, UK
$^c$ Centre for Advanced Biomedical Imaging, Department of Medicine, University College London, London, UK
$^d$ Department of Medical Education, Barts and the London School of Medicine and Dentistry, Queen Mary University of London, London, UK

** Corresponding author. Centre for Cardiovascular and Metabolic Neuroscience, Department of Neuroscience, Physiology and Pharmacology, University College London, London, UK.
* Corresponding author. William Harvey Research Institute, Barts and the London School of Medicine and Dentistry, Queen Mary University of London, London, UK.

** Corresponding author. Centre for Cardiovascular and Metabolic Neuroscience, Department of Neuroscience, Physiology & Pharmacology, University College London, London, UK.

E-mail addresses: p.hosford@qmul.ac.uk (P.S. Hosford), a.gourine@ucl.ac.uk (A.V. Gourine).

A method for simultaneous electrochemical detection of brain tissue PO$_2$ (P$_{O2}$) and pH changes together with neuronal activity using a modified form of fast cyclic voltammetry with carbon fiber electrodes is described. This technique has been developed for in vivo applications and recordings from discrete brain nuclei in experimental animals. The small size of the carbon fiber electrode ($\approx 7 \mu m$, length $\sim 100 \mu m$) ensures minimal disruption of the brain tissue and allows recordings from small brain areas. Sample rate (up to 4 Hz) is sufficient to resolve rapid changes in P$_{O2}$ and pH that follow changes in neuronal activity and metabolism. Rapid switching between current and voltage recordings allows combined electrochemical detection and monitoring of extracellular action potentials. For simultaneous electrochemical detection of P$_{O2}$ and pH, two consecutive trapezoidal voltage ramps are applied with double differential-subtraction of the background current. This enables changes in current caused by protons and oxygen to be detected separately with minimal interference between the two. The profile of P$_{O2}$ changes evoked by increases in local neuronal activity recorded using the described technique was very similar to that of blood-oxygen-level-dependent responses recorded using fMRI. This voltammetric technique can be combined with fMRI and brain vessel imaging to study the metabolic mechanisms underlying neurovascular coupling response with much greater spatial and temporal resolution than is currently possible.

** Corresponding author. William Harvey Research Institute, Barts and the London School of Medicine and Dentistry, Queen Mary University of London, London, UK.
* Corresponding author. Centre for Cardiovascular and Metabolic Neuroscience, Department of Neuroscience, Physiology and Pharmacology, University College London, London, UK.

E-mail addresses: p.hosford@qmul.ac.uk (P.S. Hosford), a.gourine@ucl.ac.uk (A.V. Gourine).

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imaging (BOLD-fMRI) (Buxton and Frank, 1997) and 2-photon excitation brain vessel imaging (Takano et al., 2006) have been widely used to study the mechanisms of the neurovascular response. However, these techniques have significant limitations. fMRI is non-invasive but suffers from relatively poor spatial and temporal resolution. This is particularly important for studies of the brain in small rodents. Moreover, the BOLD fMRI signal represents a composite response determined by changes in blood flow, blood volume and tissue oxygen consumption, and as such, lacks specificity to the underlying mechanisms that give rise to functional hyperaemia. Direct and simultaneous recordings of neuronal activity, although notably achieved during fMRI in non-human primates by Logothetis and colleagues (Logothetis et al., 2001), are not routine. Optical brain imaging is usually confined to superficial brain structures with imaging depth limited by the light scattering properties of the tissue (Helmchen and Denk, 2005). Additionally, to achieve simultaneous recordings of the vessel diameter and neuronal activity one must introduce calcium sensitive, voltage sensitive and/or cell specific dyes or genetically encoded sensors of activity. These come with their own caveats; for example there is evidence that commonly used calcium sensors introduce calcium sensitive, voltage sensitive and/or cell specific dyes or genetically encoded sensors of activity. This is particularly true for calcium imaging in vivo where calcium sensitive dyes can induce significant neural activity and thus alter the recording of interest (Bootman et al., 2018). To overcome some of these issues, we aimed to develop a minimally invasive technique for in vivo recordings of neuronal activity and associated metabolic changes with high spatial and temporal resolution.

Fast-cyclic voltammetry (FCV) using carbon fiber microelectrodes (CFM) is an established technique used to determine the dynamics of neurotransmitter release and reuptake in vitro and in vivo (Rodeberg et al., 2017). FCV relies on rapidly changing the potential of the CFM vs an Ag/AgCl reference electrode several times a second over a narrow voltage range to oxidise or reduce the analyte of interest. The amplitude of the current flux between the CFM and the analyte is recorded and used to determine the analyte concentration. Specific advantages of this technique include the small electrode size, making it ideal for studies of neurochemical changes in small brain areas. There is minimal damage to the microcirculation within the tissue and the measurements can be confined to localised brain regions located at any depth. The sub-second time resolution of FCV allows detection of events that occur at the onset of the neuronal activity and precede increases in blood flow. Separate electrochemical detection of tissue partial pressure of oxygen (P$_O_2$) and pH using FCV had been achieved previously (Bucher et al., 2014; Hosford et al., 2017; Takmakov et al., 2010). Here we describe a novel FCV-based technique that enables simultaneous recordings of key variables representing the metabolic state of the brain: brain tissue P$_O_2$, pH and neuronal activity.

Materials and methods

Carbon fiber microelectrodes

CFMs (diameter 7 μm) were made as described in detail previously (Hosford et al., 2015; Millar and Pelling, 2001). Briefly, single carbon fiber monofilaments (Goodfellow Metals) of ~10 cm in length were inserted into borosilicate glass tubing (1.5 mm O.D, Harvard Bioscience) pre-filled with acetone. After complete evaporation of the solvent, the glass tubing was transferred to a conventional horizontal micropipette puller (Model 97, Sutter Instruments) and heated to taper the glass under medium-fast pull speed. The carbon fiber bridging the two pulled electrodes was then cut and connected to a copper wire using a low-melting point tin-bismuth alloy. The carbon fiber tip was trimmed to ~80 μM in length by applying high DC voltage (~400 V), guided using a standard laboratory microscope.

Electrode calibration

All the recordings were performed using the equipment detailed in Fig. 1A. CFMs oxygen sensitivity calibration was performed in phosphate-buffered saline (PBS) containing (in mM) 137 NaCl, 2.7 KCl and 10 phosphate buffer (pH 7.4, unless otherwise stated) saturated with nitrogen to displace the dissolved oxygen. P$_O_2$ of the solution was then increased stepwise by additions of PBS saturated with 100% oxygen and monitored using an optical oxygen sensor (Oxytite®; Oxford Optronix), adjusted for temperature. A calibration curve of electrode faradaic current changes vs P$_O_2$ over a range of 5–75 mmHg was constructed. pH calibration was performed in PBS adjusted to the desired pH using additions of HCl or NaOH and monitored using a standard pH meter. Electrodes were calibrated in a small volume (3 ml) bath that allowed rapid fluid exchange. From a starting point of pH 7.4, buffer pH was changed stepwise to either 7.6, 7.2 or 7.0 in a random order for a particular electrode. A calibration curve of electrode current changes over a pH range of 7.0–7.6 was constructed.

Ethical approval and animal husbandry

All animal experiments were performed in accordance with the European Commission Directive 2010/63/EU (European Convention for the Protection of Vertebrate Animals used for Experimental and Other Scientific Purposes) and the UK Home Office (Scientific Procedures) Act (1986) with project approval from the University College London Institutional Animal Care and Use Committee. The rats were obtained from Charles River UK and housed in a temperature-controlled room on a 12 h light/dark cycle. Animals had access to standard laboratory chow and water ad libitum. On completion of the experiments, the animals were humanely killed by an anaesthetic overdose (pentobarbital sodium, 60 mg kg$^{-1}$, i.v.).

Animal preparation

Adult male Sprague-Dawley rats (280–320g) were prepared for the experiments in accord with the previously established imaging protocols (Wells et al., 2015). Anaesthesia was induced by isoflurane (2.5–4.0% in oxygen-enriched air) to establish vascular access by femoral vein cannulation. Anaesthesia was then transitioned to α-chloralose (75 mg kg$^{-1}$, i.v. initial dose; followed by supplementary doses of 10–20 mg kg$^{-1}$, i.v., as required). The right femoral artery was cannulated to monitor the arterial blood pressure. The trachea was cannulated and the animal was mechanically ventilated (~60 strokes min$^{-1}$; stroke volume - 8 ml kg$^{-1}$) with oxygen-enriched room air using a positive pressure ventilator (Harvard Apparatus) or a MR-compatible small animal ventilator (CWE).

Arterial P$_O_2$, P$_CO_2$ and pH were recorded at regular intervals using a pH/blood gas analyser (Siemens) and maintained within the physiological ranges (P$_O_2$: 100–120 mmHg, P$_CO_2$: 35–40 mmHg, and pH: 7.35–7.45) by adjusting the frequency and/or volume of mechanical ventilation. Body temperature was maintained at 37.0 ± 0.5 °C.

Fast scan cyclic voltammetry

After exposure of the skull surface by midline incision, access to the somatosensory cortex was established via a small craniotomy (~1 mm$^2$). The dura was pierced and reflected laterally to prevent damage to the microelectrode tip. Under the microscopic guidance and control of a micromanipulator, CFM was inserted into the somatosensory forelimb region of the cortex (S1FL; coordinates: 2.8–3.8 mm lateral, 1.3–1.5 mm caudal and 0.1–0.5 mm ventral from Bregma).

The CFM was slowly advanced and placed into the S1FL when clear evoked action potentials were recorded in response to the electrical stimulation (1 Hz) of the contralateral paw. The trapezoidal voltage ramps (Fig. 1B) were applied to the CFM (~4 Hz, 200 V s$^{-1}$). The CFM signals were amplified (10x), passed through a 50 Hz noise eliminator (Digitimer), filtered to 500-5000 Hz, digitised (Power1401; Cambridge Electronic Design) and recorded for offline isolation of faradaic currents corresponding to changes in pH and P$_O_2$. Continuous switching
between current and voltage recordings allowed near-simultaneous detection of the evoked potentials (voltage), pH and P\textsubscript{O\textsubscript{2}} changes (current). Electrical forelimb simulation was applied using a constant-current stimulator (Digitimer). Trains of stimulation (3 Hz, 1.5 mA, 300 μs pulse width) were applied 3 times per animal/experimental condition with intervals of at least 3 min between the stimulations. Neuronal responses were analysed by integration of the evoked volley of extracellular potentials with the baseline noise subtracted.

**Brainstem recordings**

Animals were anaesthetised and instrumented as described above. In this set of experiments the dorsal surface of the brainstem was exposed for CFM recordings as previously described (Hosford et al. 2015, 2017). The left cervical vagus nerve was exposed, separated from the sympathetic trunk and placed on the bipolar silver wire electrodes for electrical stimulation (800 μA, 1 ms at 3 Hz).

**fMRI**

fMRI was performed using a 9.4T Agilent horizontal bore scanner (Agilent) as described in detail previously (Wells et al., 2015). Briefly, a 72 mm inner diameter volume coil was used for transmission and signal was received using a 4-channel array head coil (Rapid Biomedical). To assess T2* weighted BOLD signals the following sequence parameters were used: TR = 5s, TI = 2s, matrix size = 64 × 64, FOV = 35 mm × 35 mm, TE = 10 ms, single slice (slice thickness = 2 mm), inversion pulse bandwidth = 20,000 Hz (Hosford et al., 2018). BOLD responses in the S1FL region were triggered by electrical stimulation of the contralateral forelimb.

**Results and Discussion**

**Voltammetric recordings**

Using the recording setup illustrated in Fig. 1A, two negatively-directed trapezoidal voltage ramps were applied to the CFM at an interval of 20 ms as illustrated in Fig. 1B: the first from 0 to −0.5 V, the
second from −0.5 to −1.0 V. Both voltage ramps generated a ‘background’ current due to the impedance of the electrode/fluid interface (Fig. 1Bii). The CFM currents generated during the ramps were digitised and the first was digitally subtracted from the second (Fig. 1Biii). This produced a differential background current (Fig. 1Bvi), resulting from the difference between the currents on scans 1 and 2. This subtraction procedure was used to eliminate potential non-specific signals due to changes in tissue impedance, temperature, etc. These non-specific signals appear equally on both the scans and thus are removed by subtraction, allowing discrimination of changes that occur selectively on one or the other scan. Trapezoidal ramps were applied as the background current is predominantly capacitive, reducing this current to a low level during the flat part of the trapezoid where dV/dt is zero.

Using these recording parameters, the electrodes generated distinct current profiles in response to changes in buffer pH and PO2 (Fig. 1C). Changes in [H+] caused changes in the CFM faradaic current in the voltage range between −0.1 and −0.3 V. These changes occurred only in the first of the two voltammetric scans and, therefore, were clearly present on the differential signal.

Oxygen is electrochemically reduced at voltages between −0.5 and −1.0 V, generating a cathodal faradaic current. The current from oxygen reduction appears in the signal from the second (−0.5 to −1.0 V) ramp but not from the first ramp and thus could also be seen on the differential signal. Examples of the faradaic current changes in response to changes in buffer PO2 and pH are illustrated in Fig. 1C. There is a clear separation between the two peak currents allowing detection of changes in both analytes simultaneously with minimal interference between the two. Peak current changes are shown in Fig. 1D.

**Proposed detection mechanism**

The proposed origin of the faradaic current is the pH dependent oxidation of the hydroquinone groups on the surface of the CFM (Runnels et al., 1999; Takmakov et al., 2010; Fig. 1E). Our recordings support this hypothesis as distinct double-peaks of approximately the same voltage range are detected (Fig. 1C). A third pH-dependent peak was reported by Takmakov and colleagues (Takmakov et al., 2010) and was ascribed to changes in electrode capacitance induced by protons disrupting the Helmholtz layer of charged water molecules surrounding the electrode tip and is, therefore, non-faradaic in nature. Using double-differential waveform applied during sampling, we were able to remove the effect of capacitance change (as it is present on both the waveforms) after subtraction of the resulting background current.

The reaction underlying oxygen detection is the reduction of oxygen to hydroxyl ions (Fig. 1E). This reaction occurs in a series of steps involving the formation of intermediates, including hydrogen peroxide. It is non-reversible, as evident from a single unidirectional current peak in the voltammetry scan. There have been previous proposals for the mechanism of electrochemical reduction of oxygen on carbon surfaces (Taylor and Humfray, 1975; Zimmerman and Wightman, 1991). We propose the following mechanism (Scheme 1) which includes the formation of a hydroperoxyl intermediate as well as hydrogen peroxide.

**CFM calibration**

CFM sensitivities to changes in pH and PO2 were determined by construction of standard calibration curves in the expected physiological ranges of changes in these variables, pH: 7.0–7.6 units and PO2: 5–75 mmHg. Peak faradaic current for pH detection was sampled at −200 mV on each of the scans. Current generated by oxygen was sampled once on each scan at 15 ms from the start of the flat phase of the trapezoid and at this time point the background current was minimal.

In **vitro** calibration demonstrated high CFM sensitivity to oxygen (1.1 ± 0.1 nA per 10 mmHg change in PO2; n = 10 electrodes) and protons (9.8 ± 0.8 nA per pH unit change; n = 10 electrodes). Responses were linear within the physiological ranges of changes in these variables (PO2, pH2 = 0.99; pH, R2 = 0.98) (Fig. 2A and B). CFM sensitivity to PO2 did not change over the range of physiological pH values, nor the sensitivity to protons at different PO2 levels (Fig. 2D). Current responses to a 20 mmHg change in PO2 were similar over a range of 7.0–7.6 pH units: 2.6 ± 0.2 nA at pH 7.0, 2.6 ± 0.2 nA at pH 7.2, 2.7 ± 0.2 nA at pH 7.4, and 2.8 ± 0.2 nA at pH 7.6. Current responses to 0.2 unit decreases in pH were similar over a 10–100 mmHg range of PO2 changes: 2.1 ± 0.2 nA at 10 mmHg, 2.1 ± 0.3 nA at 50 mmHg and 2.2 ± 0.3 at 100 mmHg. There was minimal interference between the two measurements; the PO2-sensitive current was altered by a mere 0.003 ± 0.005 nA per 0.1 unit pH change, while the pH-sensitive current changed by 0.1 ± 0.03 nA per 10 mmHg change in PO2 (n = 6, Fig. 2C and E).

These recordings show that there is no significant change in current produced by one analyte when the other varies over the physiological range expected in the brain tissue. Cross-talk between the proton and oxygen detection signals was calculated to be less than 5%, which is within the range reported for other electrochemical detection techniques (Tian et al., 2009). Furthermore, we can confidently exclude potential contamination of the recorded signals by other oxidizable molecules as detection of these requires a positive waveform voltage (usually +1.3 V; Park et al., 2011), while this technique records PO2 and pH changes using a negative (reducing) waveform between 0 to −1 V.

**In vivo recordings**

Using the recording setup illustrated by Fig. 3A, changes in brain PO2 were recorded during systemic hypoxia induced by a 20 s-long suspension of the mechanical ventilation. This resulted in an immediate sharp decrease in oxygen-associated faradaic current by 3.2 ± 0.6 nA, reporting a decrease in brain PO2 by 19 ± 3 mmHg (n = 6; Fig. 3Bi and ii). Upon re-instatement of lung ventilation, the brain PO2 rapidly reversed and exceeded the baseline level by 15 ± 4 mmHg within 15 s (n = 6; Fig. 3Bi and ii).

Changes in brain PO2 and pH were next recorded during systemic respiratory (hypercapnic) acidosis induced by CO2 inhalation (10% CO2 in the inspired gas mixture; 5 min). This stimulus caused a significant decrease in current by 0.31 ± 0.05 nA (Fig. 3Cii) in the voltage range corresponding to pH changes, equivalent to a decrease in brain tissue pH by 0.11 ± 0.02 units (n = 4; Fig. 3Cii). CO2-induced increase in cerebralblood flow. Upon return to normocapnia the brain PO2 decreased back to baseline within 3 min. Partial recovery of brain tissue pH was recorded during the same time period (Fig. 3Cii). The faradaic current profiles recorded in vivo during systemic hypoxia and CO2-induced acidosis were found to correspond closely to similar responses to changes in PO2 and pH recorded in vitro. Following subtraction of the control scan current from the active scan current, increases in oxygen concentration produced positive, while increases in

![Scheme 1. Proposed reaction schemes underlying the reduction of molecular oxygen on the carbon surface.](Image)
proton concentration produced negative current changes.

Electrical stimulation of the forepaw increased the neuronal activity in the S1FL of the cortex as was evident from an increase in action potential firing (recorded by the CFMs during the intervals between the applications of the voltage ramps). Traces depicted in Fig. 3D show the extracellular spike activity in S1FL, time-locked to the application of the stimulus.

Electrical stimulation of the forepaw triggered a consistent increases in faradaic current reporting significant increases in P\textsubscript{O\textsubscript{2}} (n = 6; Fig. 3E). Increases in P\textsubscript{O\textsubscript{2}} were observed 1–2 s after the onset of the stimulation. The response was found to be biphasic with an initial increase during the period of stimulation followed by a post-stimulus P\textsubscript{O\textsubscript{2}} decrease below the baseline (Fig. 3E). Calibration of the CFM after each of the recordings revealed the peak increases in P\textsubscript{O\textsubscript{2}} of 6.9 ± 1.2 mmHg and the post-stimulus decreases of 3.2 ± 2.5 mmHg (n = 6; Fig. 3E). Activation of somatosensory pathways concomitantly increased the faradaic current recorded at the sample point for the detection of pH. Changes in brain tissue extracellular pH followed a similar time course, but the pH response lagged the P\textsubscript{O\textsubscript{2}} changes by ~1 s. The pH signal displayed biphasic response profile with initial alkalinisation of 0.06 ± 0.02 pH units during the period of stimulation, followed by a decrease of 0.03 ± 0.01 pH units below baseline after the termination of the stimulus (n = 6; Fig. 3E).

There is evidence that the mechanisms of neurovascular coupling might be different in different brain areas (Devonshire et al., 2012). We next placed the CFM within the nucleus of the solitary tract (NTS) of the brainstem, and recorded changes in P\textsubscript{O\textsubscript{2}} and pH evoked by activation of visceral sensory pathways. The NTS receives mono-synaptic afferent inputs via the vagus nerve (Berthoud and Neuhuber, 2000). Electrical stimulation of the vagus nerve triggered biphasic changes in both P\textsubscript{O\textsubscript{2}} and pH in the NTS that were markedly different from those recorded in the somatosensory cortex. There was an initial decrease in P\textsubscript{O\textsubscript{2}} by 4.5 ± 2.2 mmHg and extracellular acidification by 0.095 ± 0.04 pH units, followed by a post-stimulus overshoot in P\textsubscript{O\textsubscript{2}} by 2.3 ± 0.9 mmHg with slow pHrecovery towards the baseline (n = 6; Fig. 3F).

The recordings performed within the NTS show that this technique is applicable to studies of metabolic changes within small discrete nuclei and/or regions located deep in the brain that are difficult to access using the existing imaging techniques. This could be especially useful when investigating the heterogeneity of the neurovascular coupling responses in the brain, as highlighted here by dramatically different P\textsubscript{O\textsubscript{2}} and pH response profiles recorded in the NTS (Fig. 3F) and the somatosensory cortex (Fig. 3E).

**Comparison with fMRI**

BOLD signals induced in the S1FL by activation of somatosensory pathways were recorded in identical experimental conditions to allow comparison between the profiles of the responses recorded using fMRI and P\textsubscript{O\textsubscript{2}} changes recorded using voltammetry. Electrical forepaw simulation induced biphasic BOLD signal changes in the S1FL (Fig. 4). In order to compare the BOLD signal changes with measured changes in

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**Fig. 2.** Validation of the technique and calibration of CFMs. A. Calibration curve illustrating changes in CFM electrochemical current recorded at the pH sample point in response to changes in buffer pH from 7.4 to 7.0, 7.2 and 7.6. Data points represent the means (±S.E.M) peak current changes of 10 electrodes exposed to a given change in buffer pH starting from 7.4. B. Calibration curve illustrating changes in CFM electrochemical current recorded at the oxygen sample point in response to changes in buffer P\textsubscript{O\textsubscript{2}} between 5 and 75 mmHg. Data points represent the means (±S.E.M) peak current changes of 10 electrodes, each exposed to every P\textsubscript{O\textsubscript{2}} level once. Standard linear regression was applied to fit a line for both calibrations; the equation and R\textsuperscript{2} are indicated. C. Representative traces showing changes in peak current at the P\textsubscript{O\textsubscript{2}} and pH sample points during a decrease in buffer pH of 0.2 units or increase in buffer P\textsubscript{O\textsubscript{2}} by 50 mmHg. D. Left: summary data showing peak currents generated in response to a 20 mmHg increase in P\textsubscript{O\textsubscript{2}} over a physiological range of changes in buffer pH (7.0, 7.2, 7.4 or 7.6 units). Each point represents the mean (±S.E.M) peak response of 12 electrodes. Right: summary data showing peak currents generated in response to a 0.2 unit decrease in pH over a physiological range of changes in P\textsubscript{O\textsubscript{2}} (10, 50 and 100 mmHg). Each point represents the mean (±S.E.M) peak response of 9 electrodes. E. Summary data of peak current changes recorded at the pH and P\textsubscript{O\textsubscript{2}} sample points in response to a decrease in pH by 0.2 units (left) and at the pH and P\textsubscript{O\textsubscript{2}} sample points in response to an increase in P\textsubscript{O\textsubscript{2}} by 50 mmHg (right). Shown are the means (±S.E.M) peak current responses of 6 electrodes.
The recorded voltammetry signal was down-sampled to 0.5 Hz and both signals were standardised with a z-score function and overlaid (Fig. 4 C). Plotting the distance between each sample point recorded using the two techniques revealed much of the difference observed during the post-stimulus undershoot, where it was maximal 6 s after the termination of the stimulus (Fig. 4 C).

The data obtained show that the profile of brain P_{\text{t}}O_{2} changes recorded using this voltammetric technique is very similar to the profile of BOLD responses recorded using fMRI, with an additional advantage of simultaneous detection of brain tissue pH and monitoring of neuronal activity.

**Conclusion**

Here we describe a novel experimental technique for simultaneous recordings of brain P_{\text{t}}O_{2}, pH and extracellular field potentials using CFM voltammetry. Electrochemical detection of P_{\text{t}}O_{2} and pH changes with near-simultaneous recordings of neuronal activity is possible in small nuclei located deep in the rodent brain. Simultaneous monitoring of brain blood flow (P_{\text{t}}O_{2}), metabolism (pH) and neuronal activity using the CFM-based technique described here may prove to be useful in studies of the metabolic mechanisms underlying the neurovascular coupling response.
Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

CRediT authorship contribution statement

Patrick S. Hosford: Conceptualization, Investigation, Methodology, Writing - original draft. Jack A. Wells: Investigation, Formal analysis. Isabel N. Christie: Investigation. Mark F. Lythgoe: Resources. Julian Millar: Conceptualization, Methodology, Writing - review & editing. Alexander V. Gourine: Writing - review & editing, Project administration.

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