INVESTIGATION OF A FLOWSTONE-LIKE HISTORICAL INDOOR-TRAVERTINE (RUDAS SPA, BUDAPEST, HUNGARY) USING THE $^{14}$C “BOMB-PEAK”

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ABSTRACT. Travertine precipitated during the past ca. 120 years, from thermal waters in one of the historical Spas of the Buda Thermal Karst (Hungary) was investigated using radiocarbon ($^{14}$C). The age of the deposit is based on the historic date of the structure on which the travertine was deposited. A textural study of the travertine buildup using a ~22-cm-long diamond-core crosscutting was undertaken. The original aim of the study was to improve our understanding of the controls and possibly also the rate of travertine-precipitation. In addition to characteristic, mm-scale, regular laminations, 0.5–1.0 cm dark-colored intervals were also observed in the core. Correlation of these latter textural changes with well-known changes in the water management of the Spa was greatly hindered by the lack of age data from the interior of the core. Therefore, in addition to the two known points (beginning in 1883 AD and ending 2004 AD) at least one age-datum point, somewhere inbetween, was necessary. Since the timespan of the core obviously included the 1960s of the last century, we expected that the $^{14}$C anomaly related to the atmospheric nuclear tests of those years could be detected by isotope-geochemistry. This paper gives a brief overview of the textural features of the investigated travertine and presents the dataset proving the incorporation of considerable amounts of atmospheric carbon in the carbonate precipitate, which, indeed, facilitated the indirect dating of the part of the core containing “bomb” $^{14}$C, and this helped us to unfold the factors controlling the observed textural changes of the travertine.

KEYWORDS: $^{14}$C bomb-peak, Buda Thermal Karst, historical data collection, indoor-travertine, radiocarbon.

INTRODUCTION

Budapest (capital of Hungary) is famous for its thermal springs discharging along the river Danube. The springs have been utilized for balneological purposes from the time of the Romans through Medieval ages up to the present day. They are manifestations of a large regional groundwater flow system, where the water is heated at depth and returns to the surface as thermal water, after a long “itinerary”, generally along fault zones (e.g. Kovács and Müller 1980; Mádl-Szőnyi and Tóth 2015).

Built by the Turks in the 16th century, Rudas is one of the historic spas fed by these thermal waters and has been in use from medieval times up to today. Archive documents show that in 1883 a stone-carved water-inlet was installed to the central Turkish pool of the present spa. Since then a voluminous flowstone-like travertine buildup has precipitated and covered the stand of the stone channel. Since precipitation is an ongoing process even today, the Rudas travertine was selected for detailed study with the aim of exploring the structure, texture, mineralogy and geochemistry of the material. This should contribute to our knowledge of the precipitation rate and its controlling factors in this anthropogenically influenced environment.

To attain reliable results regarding precipitation rate, in addition to the fixed date of the beginning and the end of the precipitation (see details below), it was necessary to establish at least one more precise datum preferably somewhere inbetween the two extremes. The possible incorporation of the well-known atmospheric-bomb radiocarbon ($^{14}$C) peak of the 1960s in the interior of the travertine seemed to offer such an opportunity. It was supposed
that though the thermal water was very old (11–12 kyr according to Deák 1978) and arrived to the spring along an underground pathway, it also might have been contaminated with atmospheric $^{14}$C during discharge thus contaminating the precipitate.

**GEOLOGY AND HYDROGEOLOGY**

The Rudas Spa was established at the foothills of Gellért Hill, a near-vertical rocky cliff, directly on the west side of the river Danube. The hill is built up mainly by Triassic dolomites that are deeply eroded. This karstified surface is unconformably overlain by a partly eroded late Middle Eocene transgression sequence represented by breccias, limestones, marls and clays. In Mid Pleistocene times (ca. 250 kyr to 180 kyr), travertines were precipitated due to intensely discharging thermal springs. Based on these age data, the uplift rate of Gellért Hill was calculated to be ca. 0.47 to 0.52 mm/yr (Kele et al. 2009). Recently, carbonate precipitation still occurs in active thermal springs and spas of the Gellért Hill area (cf. Virág et al. 2013a, 2013b; Kovács-Bodor et al. 2018, 2019).

In the vicinity of the Spa, there are several thermal springs discharging from the karstic reservoir which are crosscut by faults and fissures. The springwater is not of local infiltration origin but it is the result of distant regional recharge as shown by detailed hydrogeological analysis of the wider surroundings by Erőss et al. (2012), Mádl-Szőnyi and Tóth (2015), and others. Their findings are reinforced by J. Deák's $^{14}$C-based age data according to which the age of deep groundwaters of the Buda Thermal Karst is 11,000–12,000 yr (Deák 1978), i.e. they are, indeed, representatives of discharge at the end of a regional flow-path.

The average yield of the springs is relatively constant (totalling to about 3200 m$^3$/day) while both water temperature and total dissolved solids content (TDS) vary (also seasonally) from 35 to 47°C, and from 1450 to 1700 mg/L, respectively. Seasonal changes are apparently related to flooding events of the nearby river Danube, even though no advective communication between river water and groundwater has ever been detected (Schafarzik 1920; Kunszt 1947; Kessler 1965; Páll-Somogyi 2010; Mádl-Szőnyi et al. 2013).

Currently, in order to provide for a steady water supply to the Spa, instead of using the natural springs to fill the pools water is collected from deep wells and water shafts directly tapping the karstwater reservoir (see below). The water thus collected is transported by a closed pipeline under pressure first into an indoor tank and from there, through a covered channel system and according to the required amount and temperature it arrives to the pools of the Spa. Temperature is currently controlled by a thermostat system, installed in 2004.

**THE HISTORY OF RUDAS SPA AND ITS WATER SUPPLY**

Rudas Spa is one of the historic old Turkish baths of Budapest. Written records, summarized e.g. by Kunszt (1947) and Beck (1987), show that the therapeutic use of thermal springs discharging at the foothills of Gellért Hill dates back to the 13th century. Construction of the octagonal pool, the core of the present building, began only in the 16th century at the time of the Turkish occupation of Buda. Between 1880 and 1883, a major reconstruction took place with the aim of converting the traditional Turkish bath into a modern steam bath. Throughout the 120-yr-long history of this Spa, the water-supply system has been subject to several changes (see Table 1). First, it was fed by a mixture of water from five natural springs (Árpád I., Mátyás, Kinizsi, Török and Udvari Springs) with an average temperature of 42°C.
In 1926 a new spring (Gülaba, with water temperature of 42–43°C) was discovered and soon it was piped directly to the stone-carved water-inlet of the octagonal pool (Kunszt 1947). From 1926 to 1944, this served as the exclusive water-resource of Rudas. During World War II, part of the Spa was seriously damaged, but fortunately the octagonal pool survived. To optimize the utilization of thermal waters of the Gellért Hill area, a 1100-m-long water-collecting adit was driven into the dolomite in the 1970s. Thermal water has been withdrawn from the reservoir ever since then by drilled wells and shafts. Water-shaft GT-VI (water temperature between 42 and 43°C) was selected to provide the water supply to Rudas since 1988. Thermal water is pumped out of the shaft and transported by the closed pipeline under pressure first to the storage tank and from there into the pools.

As to the water management of the Spa, at the very beginning, the optimum water temperature was attained by mixing the thermal water first with riverwater, and later (up to the 2004 reconstruction) with drinking water from the rural pipelines (see Table 1). Since then, the water-management of the pools has been subject to other changes several times. Towards the very end of World War II, when the building was partially destroyed, the Spa was not used for about two years. The partly open brick-arched channel of the water-collection system for the springs collapsed and, although the ruins were cleaned up in 1945, the roof of the channel was restored only in 1947. Large-scale cleaning operations and renovations were undertaken in 1952/53, 1966/67, 1971, 1981 and 1986 (with unspecified lengths of temporary halts). During these episodes, the pool was always drained and water supply temporarily suspended. The last major alteration took place in 2004. Currently, according to the daily routine of the Spa, taps are closed for the night and the pool is drained and cleaned. The water supply is started again the next morning. The pool is filled from below, and the thermal water flowing through the stone-carved water-inlet (from which the travertine is precipitating) is maintained only for aesthetical purposes.

From archive photographs in 1883, the stone-carved water-inlet standing on the margin of the pool, did not yet have any visible encrustation on its surface (Figure 1a). In this work it was assumed that from 1883 on, the precipitation of carbonate has been quasi-continuous. Two other photographs, documenting that travertine precipitation had already occurred on the inlet-stand, originate from 1923 (Figure 1b) and from 1934. In 1968, in a photograph (made at the occasion of a movie production in the old Turkish bath of Rudas) the already fully-grown travertine buildup is clearly visible, though its color is not white but rather brownish and the size is somewhat smaller than that of today (Figure 1c).

Table 1 Water-management history of Rudas Spa.

| Years        | Water-supply                  | Temperature control |
|--------------|-------------------------------|---------------------|
| 1883–1926    | Several springs mixed (Árpád I., Mátýás, Kinizsi, Török, Udvari Springs) | Danube-water added  |
| 1926–1988 (1944–1945) | Gülaba Spring Bath out of usual function (at least for 6 months), Juventus Spring? | Tap-water added     |
| 1988–2004    | Underground water-well (GT-VI) | Tap-water added     |
| 2004 to recent | Underground water-well (GT-VI) | Thermostat          |
Figure 1  Growth history of the Rudas travertine shown by archive photographs. (a) 1883: the bare water-inlet stand with no precipitate (Photo Gy. Klötz). (b) 1923: incipient precipitate right below the inlet-channel (Photo Gy. Czvek). (c) 1968: the travertine buildup is almost totally covered by red and gray-colored Fe-Mn oxide precipitate. (d) 2012: fully developed precipitate covering the whole stand of the water-inlet. Note the red coloration (iron-oxide) along the central groove and the pure white color (CaCO$_3$) of the lateral lobes. Location of the drill-core (whole core-length) is indicated by the dashed red line. However, the 22-cm-long half-part studied section of drill-core is indicated by the continuous red line. (Please see electronic version for color figures.)
METHODS

Our work began by systematic historical data collection. In order to find eventual anthropogenic controls of travertine precipitation in the Spa available historical descriptions, reports (e.g. Molnár 1869), and the above-cited papers (Kunszt 1947 and Beck 1987) were studied, archive photographs were collected and also the management of the Spa was interviewed for information regarding daily/monthly technicalities.

In 2004, the Spa was subject to large-scale reconstruction so that in the framework of an MSc-thesis research (Veres 2007), access to the travertine body became possible. Sampling was undertaken by diamond-coring in 2004 (Bosch GDB 16000 WE portable diamond coring equipment, 2000 rev/min, water-cooled) across the widest part of the flowstone-like travertine. The 160 cm wide buildup was cored in the 235°–55° direction, at about of 5.5 cm above the pool-margin. Core recovery was 99%. To observe the internal structure, the 76.5-cm-long, 25-mm-diameter core was cut in half along its long axis (Veres 2007). Of the two roughly symmetrical, sideward-prograding, obliquely laminated parts, the more complete core slice was selected for our detailed study. The cut surface was polished, photo-documented and digitalized.

Laminae were distinguished on the basis of their color and fabric, using scanned images. Lamina counting was undertaken by starting from the initial, quasi-regular older bands of E, D and C sections ending at the anomalous section. From here on, the younger laminae (A, B) are wavy and irregular, therefore they are not appropriate for counting (Figure 2a,b). UV-fluorescence was preliminarily checked on the hand specimen (commercial UV-light source, with no filter).

Large (5 × 10 cm) thin sections (30-μm thickness) were prepared in a way to cover essentially the whole length of the half-core. They were studied by petrographic microscope (Olympus BH2) to see the fabric of the individual laminae. UV-fluorescence microscope analysis was undertaken at the Laboratories of the Geological, Geophysical and Space Science Research Group of the Hungarian Academy of Sciences with a Zeiss optical microscope equipped with a high-voltage mercury blue light-source (excitation wavelengths between 450 and 490 nm; BP) with the following filters (beam splitter: FT 510 nm and emission wavelengths from 515 nm; LP).

For X-ray powder diffraction (XRD) analyses to identify the mineralogy of the precipitate, small (ca. 5 mm) slabs were cut with a micro-saw from the core and from the most recent precipitate on the surface of the travertine buildup (the latter was taken from the pool-spar of the microterraces of the actual surface of the travertine body). The samples were ground, homogenized and analyzed with a Siemens D 5000 type diffractometer, equipped with scintillation detector. Identification of the minerals was undertaken by using the EVA software in the X-ray Laboratories of the Department of Mineralogy of Eötvös L. University, Budapest.

Trace element analyses were undertaken at the Isotope Climatology and Environmental Research Centre (ICER), Institute for Nuclear Research (Atomki), Hungarian Academy of Sciences, Debrecen, with microwave-plasma atomic-emission spectrometry, using an Agilent 4100 MP-AES instrument.

The acid-insoluble residue (recovered by (5%) acetic acid treatment of (3.5–7.8 g or ~2.5 × 1.4 to 2.4 × 0.9 cm blocks of the core) was analyzed with SEM-EDAX using the Amray 1830i scanning electron microscope of the Department of Petrology and Geochemistry of Eötvös L. University, Budapest.
14C analyses were done at HEKAL AMS Laboratory (Atomki) on 31 samples taken from the central part of the core. A 10-cm-long piece of the core sample (backward from the clearly visible layer of 1988, where the water feeding system was changed) was gradually sampled for 14C analyses with a step rate of about 0.5–1.0 cm/sample. The carbonate-powder samples were taken from the laminated travertine-core with a micro-drill. Sample preparation for accelerator mass spectrometry (AMS) analyses were conducted following standard procedures (Molnár et al. 2013a). About 10–20 mg of carbonate sample was evacuated in a two-finger reaction ampoule while there was 2 mL of 85% H₃PO₄ solution in the neighboring finger. The acid was then subsequently combined with the sample to generate CO₂. Dissolved inorganic carbon (DIC) of water samples (see Table 3) were prepared in a similar way in a septa-sealed simple reaction vessel. At first 10–20 mL of water sample was injected through a 0.45-μm pore size filter into the evacuated reaction vessel, and the 2 mL of 85% H₃PO₄ acid was injected afterwards. Acid reaction was conducted in a vacuum tight reaction vessel for at least 4 hr at 75°C temperature and the produced CO₂ gas was cleaned and separated along an on-line gas handling system of HEKAL. Finally, the clean CO₂ was graphitized.
by the sealed tube Zn-based graphitization method (Rinyu et al. 2013). The $^{14}$C/$^{12}$C and $^{13}$C/$^{12}$C ratios were measured by the EnvironMICADAS $^{14}$C AMS facility (Molnár et al. 2013b). Overall measurement uncertainty for a modern sample is $<$3‰, including normalization, background subtraction, and counting statistics. Radiocarbon results are expressed in absolute pMC units (percent Modern Carbon) (Molnár et al. 2013b).

RESULTS AND DISCUSSION

Morphology and Texture of the Rudas Flowstone-Like Travertine Buildup

The geometry of the flowstone-like travertine buildup partly follows the geometry of the underlying carved stone support, however, with remarkable anomalies. Interestingly, precipitation seems to be much less efficient in the central part where the downward flow of springwater is turbulent and is more voluminous on the flanks where flow is laminar or where springwater arrives only in the form of spray. On the actual surface of the buildup, rim-and-pool microstructures are abundant both on the near-horizontal part of the flanks and also on the quasi vertical, slightly concave central face. The concave central face is peculiar because of its deep red color implying Fe-oxide precipitation and concomitant development of acidity (~ferrolysis sensu Brinkman 1970 or Van Breemen 1988 and Loeppert 1988). This acidity also explains the suppression of carbonate precipitation in this zone and the resulted concave morphology of the central part of the build-up. The flowstone-like travertine has two large more-or-less symmetrical lobes (Figure 1d).

Diamond coring revealed that the interior of the flowstone is built up by alternating light-colored porous macrocrystalline and dark-colored, grayish dense “micritic” laminae, the frequency, thickness and regularity of which show substantial changes along the total 76.5-cm-long core. Since the two sides, representing the two lobes of the build-up, are quasi symmetrical, in the followings description, only of the left-side lobe will be given in detail. The more complete half-part section studied represents 22 cm of the whole core-length (Figure 1d). Closer scrutiny of the sections crosscutting the flanks (or side-lobes) showed that further subdivision of the lamination was possible (Figure 2a): (E) regular, fine prograding lamination with gradually increasing dip (from 30–70° away from the center), (C and D) wider and less sharply developed, more porous laminae, (A and B) varicolored, irregular to wavy laminae with a predominance of porous crystalline material. In each section, occasional subtle disconformities or “growth bands” could be detected. The boundary between section (B) and (C) is marked by an about 2-cm-thick dark-colored, strongly UV-fluorescent intercalation, different from both the underlying and the overlying laminae (Figure 2b). Based on results of lamina counting, 23 regular lamina-pairs were detected in the Section E (13–20 cm in the core.) Then 10 lamina-pairs were counted in Section D (8.7–13 cm) and 20 lamina-pairs were seen in Section C (5.5–8.7 cm), however, these latter are less sharply developed. Because of the wavy (rim-and-pool structured) laminae are not properly visible in Sections A and B, they were not counted (see Figure 2a,b).

Evaluation of the observed textural features (Figure 2a)—also considering the major technical changes documented during the history of the spa—suggested that the first, regularly prograding Section (E) should be the product of the first historical period of the spa when only natural spring water (Árpád I., Mátyás, Kinizsi, Török and Udvari Springs) entered the pool thorough the inlet (1883–1926). The (C and D) Sections characterized by thicker, more porous laminae would then be representative of a longer period, when only one particular spring (Gülbaba Spring) was used as a feeder to the pool (1926–1988). At the
boundary of Sections (C) and (D), probably in the middle of the 1940s, for some period, the water of another spring (probably that of the higher temperature Juventus) was also utilized to feed the octagonal pool.

An additional change occurred between Sections (B) and (C), namely a ca. 1-cm-thick, darker colored layer. $^{14}$C-analyses of samples of the core including this particular group of layers clearly showed that the $^{14}$C-maximum (bomb-peak, see below) corresponded exactly with this dark-colored zone between Sections (C) and (B). With the position of the mid-1960s thus fixed, it was already possible to assign the textural changes of Sections (C) and (D) to those changes of the water-management which were recorded in the archives, as already discussed. Also, it became clear why the succeeding wavy lamination and the predominance of porous macrocrystals of calcite are so characteristic of the the last Sections (A and B). These probably reflect the beginnings of the formation of the actual rim-and-pool microstructures clearly seen on the surface of the buildup. It appears that the boundary between Sections (A) and (B) also reflects the change in the water-management of the Spa, where from 1988 onwards, thermal water is supplied exclusively from water-shaft GT-VI) (Figure 2a).

The regular, high-frequency “background” lamination on to which all antropogenically controlled lower-frequency variations are superimposed, are apparently of natural hydrological origin (Virág et al. 2013b) and will be addressed in details in another paper.

Mineralogy, Micro-Structure, Micro-Texture and Geochemistry

X-ray powder diffractograms showed that the bulk of the Rudas travertine is made up of low-Mg calcite ($\text{MgCO}_3$ mol\% 2–4). Only the most recent precipitate on its surface (the coarse-grained pool-spar of the microterraces) is high-Mg calcite with 8 mol \% $\text{MgCO}_3$. No aragonite was detected. Preliminary study of the acid-insoluble residue by SEM-EDAX showed the presence of clay (e.g. smectite, kaolinite) and fine silt-size detrital minerals (quartz, rutile) some U- and Th-bearing grains, barite, Fe-oxides with or without Ti, and few Cu, Zn-rich grains. The presence of S and Cl was also detected.

Micropetrographic studies revealed the internal structure and texture of the layers of the travertine core. The dense laminae consist of thin dark-brown colored “micritic”/microsparry material, while the porous laminae consist of coarse dendritic calcite crystals (similarly to the fabric described by Jones 2005, Jones and Renaut 2010 from hot-spring travertines).

UV fluorescence of the hand-specimens is correlated with the laminated structure of the travertine. As a rule, dense compact laminae proved to be more fluorescent than their porous counterparts. Red-colored, Fe-oxide-stained bands are generally dark or only very weakly fluorescent while the fluorescence of the 0.5-cm-thick brownish-colored “anomalous” couplet at the boundary of Sections B and C (at 5–5.5 cm) is very strong (Figure 2a,b).

Between 4.5 to 5.5 cm, the relatively compact dark brown-colored laminae showed the most intense fluorescence at 5–5.5 cm whereas the successive compact laminae (mostly the gray and red-colored ones) showed no fluorescence at all (Figure 2c,d).

The concentration of trace elements is likewise anomalous in this particular section of the core. In the case of the sample at 5.3 cm there, is a significant enrichment in Mg, Sr, Al, Na and K. On the other hand, at 4.9 cm, 4.7 cm and 4.4 cm there is a depletion in Mg, Sr and Na, however,
this is accompanied here by an enrichment in Fe and Mn (at 4.7 cm), Cu (at 4.9 cm and at 4.7 cm). At 4.7 cm, there is a slight enrichment in Al and Na when compared to sample at 4.9 cm and a very little enrichment in K when compared to that at 4.4 cm (Figure 3, Table 2). The concentration of Ba at 21.0 cm reaches 50 mg/kg (RT-1), however, in the other samples the Ba content is negligible.

**Discussion of the Observed Mineralogical, Geochemical and Textural Features**

The presence of Al, K and Fe can be related probably to the non-carbonate fraction mentioned above. Based on her studies of a large Pleistocene travertine complex at Süttő (NW of Budapest) Török (2018) and Török et al. (2019) suggested that the majority of such trace elements were of detrital origin and they accumulated on the travertine surface as dust at times of temporary subaerial exposure. Temporary halts of the water-supply might have also resulted in dust (aerosol)-deposition on the surface of the Rudas travertine buildup, thus enhancing the non-carbonatic contamination of the subsequently precipitating travertine layer. However, Ca, Mg, Na, Ba, S and Sr are apparently related to the carbonate fraction (cf. Pentecost 2005). Likewise, Mg, Sr and Ba can be incorporated in the calcite crystal lattice for substitution of Ca (Fairchild and Treble 2009), Na is known to preferentially enter the interstitial space of the crystal lattice. The relatively high concentrations of Mn and Fe probably originate from co-precipitated Mn- and Fe-oxides or hydroxides.

Similar results were published by Kovács-Pálffy and Földvári (2004) and Korpás et al. (2004) on Pleistocene travertines from the NE Transdanubian Range, among those from the Buda Castle Hill (about 600 meters from the Rudas Spa). In the insoluble residue of the travertines of the Castle Hill illite/montmorillonite (or smectite), illite, kaolinite, barite, hematite, and also some, quartz, feldspar, muscovite, and chlorite were detected.

Fluorescence of carbonates may result (or be quenched) by various trace metals substituting in the crystal lattice of carbonate minerals (e.g. McGarry and Baker 2000, White and Brennan 1989), or alternatively by residual organic matter preserved as inclusions in the carbonate phase (Dravis and Yurewicz 1985, Fairchild and Baker 2012). The presence of metallic ions, Fe, Mn and Pb, quenches fluorescence, but Al may either quench or increase intensity. Increased copper concentrations also decrease fluorescence. These trace element concentrations within the analyzed samples may—at least partially—explain the strong fluorescence at 5–5.5 cm followed by very weakly fluorescent gray- and red-colored, Mn, Fe and Cu enriched layers at 4.5–5 cm in the travertine-core.

However, the reason for the intensely fluorescent dark-brown micritic zone between 5.0 and 5.5 cm may rather be the presence of incorporated organic matter in the carbonate phase. In travertine-like precipitates, copious organic activity is the norm rather than the exception and its effect may be to increase fluorescence, as suggested by Arp et al. (2010).

As already mentioned, the peculiar red color and the concave (dissolutional) morphology of the central part of the buildap are expected to have been the results of ferrolysis promoted by oxygenation and hydrolysis of dissolved Fe^{2+} species in the water. In this central part of the travertine buildup, not only turbulence, CO₂-degassing, but also aeration/oxygenation and ferrolysis are apparently intense, the latter promoting dissolution. On-site observations suggested that microbial activity may also be involved in these processes. At times of large scale cleaning and renovation works in the Spa (as in 1966–1967), the pool was always drained and water supply temporarily suspended. After these works, the newly precipitating
Figure 3  Distribution of trace element concentrations along the studied half-section of the travertine-core (see also Table 2). (a) Macropetrography of the core across the left lobe of the buildup (sampling points for $^{14}$C are indicated by dark blue dots; subsamples for trace element analyses are shown by blue squares; $^{14}$C subsamples are indicated by dark blue dashed line). For the details of the subsamples, see Figure 4). (b) Mg, Sr, Fe, Mn, Al, Na, K and Cu trace elements (mg/kg or ppm) are shown by blue bars while those occurring in the "bomb-peak" section (at 5.3 cm) are marked by red bars (samples are displayed in the order of their cm distance from the "recent" end of the core, yr 2004).
Table 2 Trace element contents of the Rudas travertine (samples are displayed in the order of their cm distance from the “recent” end of the core, yr 2004). Italics indicate sampling points measured also for $^{14}$C (see Table 3).

| Distance from the recent top (cm) | Sample code | Mg (mg/kg) | Sr (mg/kg) | Fe (mg/kg) | Mn (mg/kg) | Al (mg/kg) | Na (mg/kg) | K (mg/kg) | Cu (mg/kg) |
|----------------------------------|-------------|------------|------------|------------|------------|------------|------------|-----------|------------|
|                                  |             | (ppm)      | (ppm)      | (ppm)      | (ppm)      | (ppm)      | (ppm)      | (ppm)     | (ppm)      |
| 0.1                              | RT-8        | 7563       | 1349       | 106        | 60         | 0          | 608        | 73        | 0          |
| 4.1                              | RT-7        | 6609       | 1556       | 468        | 118        | 111        | 678        | 133       | 51         |
| 4.4                              | RT-6        | 6772       | 1466       | 573        | 178        | 37         | 628        | 201       | 72         |
| 4.7                              | RT-5        | 6143       | 1543       | 3917       | 282        | 164        | 710        | 218       | 413        |
| 4.9                              | RT-4        | 6250       | 1617       | 536        | 174        | 114        | 681        | 183       | 506        |
| 5.3                              | RT-3        | 8282       | 1803       | 306        | 81         | 193        | 813        | 238       | <25        |
| 9.5                              | RT-2        | 6754       | 1455       | 223        | 140        | 136        | 633        | 145       | 0          |
| 21.0                             | RT-1        | 6125       | 1554       | 244        | 213        | 37         | 600        | 51        | 60         |

Elemental composition of travertines (in Pentecost 2005)

| Metogene (ppm)                  | Termogene (ppm) | Average limestone (ppm) (Turekian & Wedepohl 1961) |
|---------------------------------|-----------------|-----------------------------------------------------|
| Elemental composition           |                 |                                                     |
| Metogene (ppm)                 |                 |                                                     |
| Elemental composition of travertines (ppm) (in Pentecost 2005) | 60 to 35000     | 47000                                               |
| Termogene (ppm)                | 20 to 14000     | 610                                                 |
| Average limestone (ppm) (Turekian & Wedepohl 1961) | 8 to 37000      | 3800                                                |
travertine turned generally red- and/or gray-colored (as illustrated on the 1968 photograph). These colors appear to be preserved as episodic, darker-colored laminae in the interior of the travertine buildup. A colored band is particularly apparent in the core at ~5 cm. Comparing to the other sections, this band is relatively rich in Fe$^{3+}$ (3917 mg/kg) and Mn$^{2+}$ (3917 mg/kg) at 4.7 cm, as well as in Cu (506 mg/kg and 413 mg/kg at 4.9 cm and 4.7 cm, respectively) (Table 2, Figure 2a, Figure 3).

Since microbiological analyses of other springs of the Gellért Hill by Borsodi et al. (2012, 2018), Anda et al. (2014), Makk et al. (2016) and Enyedi et al. (2019) proved the presence of various bacterial communities, the role of preserved microbial biofilms in concentrating a range of trace elements in the Rudas travertine should also be taken into consideration. The elevated natural radioactivity also is apparently connected to biofilms, since biogeochemical precipitates readily adsorb $^{226}$Ra and produce $^{222}$Rn, which although short lived might be a problem when air quality is considered (Erőss et al. 2012; Kovács-Bodor et al. 2019).

**Radiocarbon Results of the Rudas Travertine**

The atmospheric bomb-$^{14}$C peak gave a unique possibility to identify the years of 1960s within the studied travertine core. If we can detect the incorporation of $^{14}$C in the 1960s travertine-layers that would also be an indication of the possible admixture of atmospheric/freshwater carbon to the discharging thermal water modifying the C-isotope signal of the precipitated carbonate phase.

There are several possibilities how atmospheric carbon might contaminate the carbon pool of the thermal water that arrived gravitationally from the springs to the spa via a relatively long (~400 m) and partly open brick-arched channel system up till 1988. As mentioned above, since 1988, thermal water is pumped out of the GT-VI shaft and transported by a closed pipeline under pressure first to a storage tank and from there into the pool. Up to 2004, the temperature of the spring-water was adjusted in the spa by adding colder fresh water directly from the river and later tap-water from rural pipelines that also originated from river Danube (see Table 1). Since these are both mainly of rainwater origin, it is to be expected that they would be contaminated by the bomb-peak $^{14}$C in the 1960s. Although the dominant DIC fraction in the spa water arrives directly from the thermal-water reservoir, during precipitation on the stone-carved water-inlet, there could easily be partial exchange with atmospheric CO$_2$, especially due to the turbulent water-flow. This atmospheric contamination process could also contribute to the atmospheric $^{14}$C bomb-peak presence in the precipitated carbonate layers (cf. travertines). An atmospheric contribution was clearly detected when we measured and compared the $^{14}$C signal of recent incoming water DIC and freshly precipitated carbonate layers. The difference was about +2.6 pMC (from water DIC to the freshly precipitated carbonate), which indicates a significant atmospheric C contribution during the precipitation process.

The DIC content of the original thermal spring water, sampled directly at the source well was analysed for $^{14}$C by LSC techniques (by Deák and Süveges, in Deák 1978). The results are also given in Table 3.

In Table 4, the average mean of the $^{14}$C activity of the investigated Rudas travertine samples was 8.8 ± 2.0 pMC (n=13). However, in the middle of the 10-cm-long analyzed core (Figure 2a,b, Figure 4) we identified a $^{14}$C increase (21.39 ± 0.10 pMC) at 5 cm which could be the signal of the atmospheric $^{14}$C bomb-peak (around AD 1963) (Figure 4, Tables 3 and 4). To get better resolved analyses around the observed $^{14}$C increase, we have resampled this location
Table 3  $^{14}$C pMC results of the analysed samples of the Rudas drill-core, the recent thermal spring and spa-water and its recently precipitated carbonate. Italics indicate sampling points also measured for trace element analyses (see Table 2). The increased $^{14}$C values and related samples are highlighted by bold numbers.

| AMS $^{14}$C lab code | Sample code | Distance from the recent (2004) top (cm) | $^{14}$C pMC (%) | pMC unc. (%) (1s) |
|-----------------------|-------------|-----------------------------------------|----------------|-------------------|
| DeA-2522 RC-1         | 1.2         | Section A                               | 7.77           | 0.06              |
| DeA-2535 RC-2         | 1.9         |                                         | 13.11          | 0.07              |
| DeA-2523 RC-3         | 2.2         |                                         | 10.29          | 0.07              |
| DeA-2534 RC-4         | 2.9         | Section B                               | 7.46           | 0.06              |
| DeA-2524 RC-5         | 3.4         |                                         | 10.16          | 0.07              |
| DeA-2533 RC-6         | 4.1         |                                         | 9.14           | 0.06              |
| DeA-2525 RC-7         | 4.4         | Anomalous                               | 7.35           | 0.06              |
| DeA-2526 RC-8         | 5.3         | Section                                  | 21.39          | 0.10              |
| DeA-2527 RC-9         | 6.1         |                                         | 8.79           | 0.06              |
| DeA-2528 RC-10        | 7.0         | Section C                               | 11.37          | 0.07              |
| DeA-2529 RC-11        | 7.9         |                                         | 9.20           | 0.07              |
| DeA-2530 RC-12        | 8.9         |                                         | 7.05           | 0.06              |
| DeA-2531 RC-13        | 9.9         | Section D                               | 6.74           | 0.06              |
| DeA-2532 RC-14        | 11.0        |                                         | 6.13           | 0.06              |
| DeA-1834 RC-O         | 3.8         |                                         | 9.18           | 0.10              |
| DeA-1835 RC-N         | 4.0         |                                         | 9.11           | 0.09              |
| DeA-1836 RC-M         | 4.2         |                                         | 9.49           | 0.09              |
| DeA-1837 RC-L2        | 4.4         |                                         | 8.59           | 0.08              |
| DeA-1838 RC-L1        | 4.5         |                                         | 9.93           | 0.09              |
| DeA-1839 RC-K         | 4.6         |                                         | 9.10           | 0.10              |
| DeA-1840 RC-J         | 4.8         |                                         | 7.22           | 0.07              |
| DeA-1841 RC-I         | 5.0         |                                         | 8.30           | 0.08              |
| DeA-1842 RC-H         | 5.2         |                                         | 20.46          | 0.14              |
| DeA-1843 RC-G         | 5.4         |                                         | 23.17          | 0.14              |
| DeA-1844 RC-F         | 5.5         |                                         | 22.54          | 0.15              |
| DeA-1845 RC-E2        | 5.6         |                                         | 12.57          | 0.10              |
| DeA-1846 RC-E1        | 5.7         |                                         | 10.15          | 0.09              |
| DeA-1847 RC-D         | 5.8         |                                         | 7.23           | 0.08              |
| DeA-1848 RC-C         | 6.0         |                                         | 9.23           | 0.09              |
| DeA-1849 RC-B         | 6.2         |                                         | 12.49          | 0.10              |
| DeA-1850 RC-A         | 6.4         |                                         | 9.36           | 0.09              |

Table 4  Higher resolution $^{14}$C analyses in the core. The increased $^{14}$C values and related samples highlighted by bold numbers.

| AMS $^{14}$C lab code | Sample code | Distance from the recent (2004) top (cm) | $^{14}$C pMC (%) | pMC unc. (%) (1s) |
|-----------------------|-------------|-----------------------------------------|----------------|-------------------|
| DeA-15544 RC-O        | 3.8         |                                         | 9.18           | 0.10              |
| DeA-15543 RC-N        | 4.0         |                                         | 9.11           | 0.09              |
| DeA-15542 RC-M        | 4.2         |                                         | 9.49           | 0.09              |
| DeA-15541 RC-L2       | 4.4         |                                         | 8.59           | 0.08              |
| DeA-15540 RC-L1       | 4.5         |                                         | 9.93           | 0.09              |
| DeA-14865 RC-K        | 4.6         |                                         | 9.10           | 0.10              |
| DeA-14864 RC-J        | 4.8         |                                         | 7.22           | 0.07              |
| DeA-15252 RC-I        | 5.0         |                                         | 8.30           | 0.08              |
| DeA-14863 RC-H        | 5.2         |                                         | 20.46          | 0.14              |
| DeA-14862 RC-G        | 5.4         |                                         | 23.17          | 0.14              |
| DeA-14861 RC-F        | 5.5         |                                         | 22.54          | 0.15              |
| DeA-16560 RC-E2       | 5.6         |                                         | 12.57          | 0.10              |
| DeA-14860 RC-E1       | 5.7         |                                         | 10.15          | 0.09              |
| DeA-16558 RC-D        | 5.8         |                                         | 7.23           | 0.08              |
| DeA-14858 RC-C        | 6.0         |                                         | 9.23           | 0.09              |
| DeA-14857 RC-B        | 6.2         |                                         | 12.49          | 0.10              |
| DeA-14856 RC-A        | 6.4         |                                         | 9.36           | 0.09              |
Figure 4  Distribution of $^{14}$C pMC (%) along the 12-cm-long section of travertine-core (the cm-scale starts with zero at the “recent” end of the core (2004); see also Tables 3 and 4). (a) Macropetrography of the core across the 12-cm-long section from the left lobe of the buildup (sampling points for $^{14}$C are indicated by dark blue dots; $^{14}$C subsamples are indicated by red rectangles; $^{14}$C (sub-)samples taken from the zone of the “bomb-peak” are highlighted with bold. (b) Distribution of $^{14}$C pMC (%) along the full 12-cm-long studied section of travertine-core. Red dashed lines indicate the position of the detected $^{14}$C anomaly. The anomaly coincides with the marker horizon at 5.3 cm (see also Table 3). Blue dashed lines indicate the section studied in detail (RC-A to RC-O samples) from 3.8 cm to 6.4 cm of travertine-core (c). Location of the peak is highlighted by red columns and red dashed lines from 5.2 cm to 5.5 cm (see also Table 4).
in 0.1–0.2-mm steps from 3.8 cm to 6.4 cm of the core (Table 4). The majority of the samples (n=11) shows again the average 8.9 pMC mean value, but we can still observe a peak over 20 pMC.

As the amplitude of this $^{14}$C peak (about 21–23 pMC) in the travertine is close to double the level of $^{14}$C in the other layers, it can be correlated with the doubling of the atmospheric $^{14}$C level during the bomb-peak (about 200 pMC in the year of 1963) (Hua and Barbetti 2004; Hua et al. 2013).

CONCLUSIONS

A flowstone-like travertine buildup precipitated on the surface of the stone-carved water-inlet of the central Turkish pool of the Rudas Spa recorded the past ca. 120 years history of natural and anthropogenic processes affecting the Spa. In order to understand the details of these processes (and their imprints on the travertine), sampling was undertaken by diamond-coring and the core was cut into half along its long axis and analysed by the above-described petrographical and geochemical methods. Despite the fact that the Spa is fed by old (Pleistocene) thermal-waters arriving from a deep-seated reservoir, radiocarbon analyses proved to be useful in improving the dating of the core. A $^{14}$C peak (23.17 ± 0.14 pMC), which we identify as related to the radiocarbon bomb-peak, resulted from the atmospheric nuclear tests of the early 1960s was successfully identified in the travertine. This exact datum at about the middle of the core made possible to correlate observed textural changes of the travertine with archive documents of the decadal modifications in the water-management of the Spa.

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