Drug Delivery System for Controlled Cancer Therapy Using Physico-Chemically Stabilized Bioconjugated Gold Nanoparticles Synthesized from Marine Macroalgae, *Padina Gymnospora*

Manoj Singh1,2, Manish Kumar2, S Manikandan1, N Chandrasekaran2, Amitava Mukherjee1 and AK Kumaraguru3

1Department of Marine and Coastal Studies, Madurai Kamaraj University, Tamilnadu, India
2Advanced Instrumentation Research Facility, Jawaharlal Nehru University, New Delhi, India
3Centre for Nanobiotechnology, VIT University, Tamilnadu, India

Abstract

Gold nanoparticles (AuNPs) have always been considered as superior tools for biosystem applications owing to their characteristic optical properties in form of surface plasmon resonance and amperometric properties with very low or no immediate toxicity. Biosystem application based tools have been designed based on many studies with gold nanoparticles but the reduction of bulk material to nanosized particle in conjugation with biomolecules in a physico-chemical environment is an area requiring deeper investigation. In the present study complementary high resolution imaging techniques on different length scale are applied to elucidate morphology of gold nanoparticles. The biomolecules involved in conjugation and reduction were further characterized. The impact of macroalgae broth concentration on formation of AuNPs (8-21 nm) were further studied to determine the functional and molecular mechanism of cell death on Liver cancer (HepG2) cell line and Lung cancer (A549) cell line. We report AuNPs-induced death response in human carcinoma liver cell line HepG2 in contrast to lung cancer cell line which remained little affected. The induction specificity for death response in lung cells clarifies that AuNPs do not universally target all cell types. Altered DNA fragmentation and cell staining in different cancer cell suggests a potential for in vivo applications of gold nanomaterials and demands the need of the time for studies evaluating the interactions of nanomaterials with biomolecules and cellular components for controlled cancer therapy.

Keywords: Cellular effects; Physico-chemical parameter; Drug delivery system; Amperometric; Gold nanoparticle

Introduction

Nanoscale materials hold great promise for both industrial and biomedical applications. Since, cancer is observed as the most dangerous class of disease categorized by uncontrolled cell growth [1]. Cancer is the third leading cause of death (after heart disease and stroke) in developed countries and the second leading cause of death (after heart disease) in the United States (see http://www.cdc.gov). Lung, stomach, liver, colon and breast tumour are the main types of tumors that lead to total death worldwide. Studies have shown that there were 10 million new cases, 6 million deaths, and 22 million people living with cancer worldwide in the year 2000 [2]. These numbers represent an increase of about 22% in incidence and mortality from that of the year 1990. It is projected that the number of new cases of all cancers worldwide will be 12.3 and 15.4 million in the year 2010 and 2020, respectively [3]. In many types of cancer, we are yet to find a satisfactory medicine or carrier of medicine as in case of drug delivery to be used as a satisfactory chemotherapeutic agent [4].

Thus, nanotechnology, an interdisciplinary research field comprising chemistry, engineering, biology, and medicine, has great potential for early detection, accurate diagnosis, and tailored treatment of cancer [1]. Toxicological studies suggest that nanoparticles may cause adverse health effects, but the fundamental cause–effect relationships are ill defined. Thus, the interaction of nanoparticles with biological systems including living cells has become one of the most urgent areas of collaborative research in materials science and biology [5]. The most interesting properties of nanoparticles, that is, the quantum size effect or surface-induced effects, result from their minute size. Nanoparticles are of similar size to typical cellular components and proteins, and thus may bypass natural mechanical barriers, possibly leading to adverse tissue reaction. These nanoparticles can offer unprecedented interactions with biomolecules both on the surface and inside the body cells, which may bring revolution in cancer diagnosis and treatment [6]. The well-studied nanoparticles for targeting cancer include paramagnetic nanoparticles [7], liposome [8] and gold nanoparticles [9]. In addition, AuNPs also serve as specialized microscopic probes to study cancer cells, because AuNPs selectively accumulate in tumor cells, showing bright scattering [10].

Thus, there is an increasing need to develop high-yield, low cost effective, nontoxic and environmentally benign procedures for synthesis of metallic gold nanoparticles. Therefore, the biological approach for synthesis of nanoparticles becomes important. The potential of plants as biological materials for the synthesis of nanoparticles is yet to be fully explored. The importance of seaweeds as a marine resource for biomimetic synthesis of metallic nanoparticle may be emphasized due to the increasing demand for medicinal products like antibiotics such as antibacterial, antifungal, antiviral and antitumour activity with enhanced properties. From the above experiments it is also known that green synthesis showed high stability under the preparatory conditions. These results prompted us to start a series of cell-based experiments, in which the bioconjugated gold nanoparticles might show differential permeability for human cancer cells.

*Corresponding author: Manoj Singh, Department of Marine and Coastal Studies, Madurai Kamaraj University, Madurai, Tamil Nadu, India, Tel: 09994796962, E-mail: marinenanomed@gmail.com or Manoj.Singh@vit.ac.in*

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Material and Methods

Bioreduction and surface characterization gold nanoparticles (AuNPs)

Bioreduction of bulk gold chloride to synthesize AuNPs was conducted with 10⁻⁴M trivalent aurum solutions with different broth concentration (1, 3, 5, 10 and 20%) of brown seaweed (Padina gymnospora) leaf extract (SLE). The effect of different leaf broth concentration was investigated for the optimum and controlled synthesis of gold nanoparticles. The optimum pH of 9.5 and 75°C temperature of the reaction medium was kept constant and the absorbance values of broth solutions were measured between 300-700 nm spectrophotometrically. The process of separation of biological debris from nanoparticles was established following centrifugation and redispersion of broth solution in sterile deionized water thrice at 12000 rpm for 20 minutes to obtain better separation of entities from the nanoparticles. The pellet obtained at final centrifugation was frozen at -70°C to remove water residue and lyophilized using Lyophillizer (Micro Modulyo 230 freeze dryer, Thermo Electron Corporation, India). The lyophilized nanoparticle samples were stored at 4°C for future cytotoxicity study. The lyophilized nanoparticle samples were stored at 4°C [11]. HRTEM (JEOL TEM-2100F) & Energy dispersive X-ray analysis (EDX) was applied to determine the surface morphology, size pattern and the chemical composition of the AuNPs by drop coating of dried gold nano powder on to carbon film at an operating voltage of 200 kV.

Cell culture and cytotoxicity assays

Two cell lines, Human hepatocellular liver carcinoma; HepG2 and Human lung adenocarcinoma epithelial cell line; A549 were used in this study. All cells were cultured in Dulbecco's modified Eagle medium (DMEM) at 37°C in water-saturated air supplemented with 5% CO₂. Culture media was changed and passaged once in three days. Cell numbers were estimated using a cell counter (Schaerfe cell counting chamber). Culture dishes were gently washed twice with MEM and then the monolayer of cells in the dishes were exposed to 1ml of MEM medium containing defined different concentration of the gold nanoparticles. Control cells were treated with 0.25% DMSO. After 48h of culture, cellular DNA was extracted by proteinase K digestion [14] from treated and untreated tumour cells. 1×10⁶ cells were re-suspended in 1ml of cell lysis buffer containing 50 mM Tris HCl (pH 8.0), 10 mM ethylenediaminetetraacetic acid (EDTA), 0.1 M NaCl, and 0.5% sodium dodecyl sulfate. The lysate was incubated with 0.25 mg/ml RNase A at 37°C for 30 minutes, and then with 0.2 mg/ml proteinase K at 50°C overnight. DNA was extracted by phenol: chloroform: isoamyl alcohol (25:24:1) mixture, precipitated in ethanol. DNA electrophoresis was performed in 1% agarose gel containing ethidium bromide (0.6 μg/ml) at 70 V, and the DNA fragments were visualized and photographed under a UV transilluminator.

Morphological changes of apoptosis

Cells undergoing apoptosis display typical condensed morphology, namely cell shrinkage, chromatin condensation and nuclear fragmentation. Dramatic changes occur within the nucleus during apoptotic death. In addition, patches of localised partially condensed chromatin were found abutted along the inner part of the nuclear membrane. Chromatin condensation, nuclear shrinkage and formation of apoptotic bodies can easily be observed under confocal laser scanning microscopy, after appropriate staining of nuclei with DNA-specific fluorochromes. V-FITC Kit was used for detection and discrimination of apoptotic, necrotic and dead cells [15], chromatin condensation was performed by Hoechst staining [16] while confirmation of apoptosis was achieved using Acridine orange Staining [17].

Statistical analysis

Lethal concentration dose for growth inhibition were derived using Probit software (EMSL-Cincinnati) and graphs were computed using Graph Pad Prism and Origin Pro software.

Results and Discussion

The light absorption pattern of different macroalgae biomass concentration with 1 mM HAuCl₄ solution was kinetically monitored with the help of UV-Vis measurement at regular time interval for formation of Au nanoparticles (Figure 1). The presence of SPR peak between 500-550 nm for each broth concentration clearly suggests the formation of gold nanoparticles, which is the characteristic of gold nanoparticles in solution. The increase in absorbance was seen as a function of reaction time. The most stable peak was observed at 544 nm for macroalgae Padina gymnospora at 5 g aqueous leaf extract. Flat gold nanoparticles absorb the wavelength in the NIR region of the electromagnetic spectrum, which corresponds to the longitudinal surface plasmon absorption [18]. The stable synthesis of gold nanoparticle with narrow size distribution was obtained which was further confirmed through Transmission electron microscopy.

Effect of broth concentrations

The gold nanoparticle synthesized from 1mM solution of HAuCl₄ and 5% aqueous leaves extract of Padina gymnospora at constant temperature (75°C) and pH (9.5) was further investigated for the possibility of controlling the particle size and shape by changing the composition of the reaction mixture. It was found that rate of synthesis wasn’t realized at low macroalgae concentrations. Probably, reductant ions couldn’t be high enough to reduce gold ions [19]. Au nanoparticles of different sizes (8 to 21nm) were obtained at different concentration of macroalgae (1, 3, 5, 10 and 20%) broth by keeping control of other
The particle size was observed to decrease with an increase in the leaf broth concentration. The optimum concentration of macroalgae leaf extract reacting with HAuCl₄ led to the formation of spherical nanoparticles with diameter ranging from 8 to 21 nm (Figure 3D), while the large quantities of mixture nanoparticles were formed on increasing the concentration of the Padina gymnospora leaf extract (Figure 3C). Large quantities of small size nanoparticle with the average diameter of 13 nm were formed. The high-resolution TEM (HRTEM) images displayed clear lattice fringes on the particle surfaces. Selected area electron diffraction pattern (SAED) of a single spherical particle confirmed the single crystalline nature of gold nanoparticles with the fcc phase. Scherrer ring pattern corresponding to (1 1 1), (2 0 0) and (2 2 0) planes of the fcc crystalline lattice of gold was observed for each sample (Figure 3E). Similar changes in the shapes and size were observed on phyllanthin assisted gold nanoparticle biosynthesis [19]. The use of a low concentration of the extract reacting with gold chloride led to the formation of hexagonal or triangular gold nanoparticles, while the shape of the nanoparticles changed to spherical on increasing the concentration of the macroalgae leaves extract. Song et al. [21] has recently reported that control of the shape and size of metallic nanoparticles is very much required to enable the tuning of physico-chemical parameters (Figure 2). The statements of size used in the following relate to the diameter of the gold core. The low concentration of macroalgae biomass with HAuCl₄ led to the formation of increased absorbance. As soon as the broth concentration was increased beyond 20%, the solution started forming aggregates resulting in poor synthesis of AuNPs. From the result, it was concluded that the optimum absorbance occurred at 5% of macroalgae biomass. The results were similar with Shipway et al. [20] who used sodium citrate to reduce gold chloride for gold nanoparticle synthesis and he concluded that amount of reductant (sodium citrate) is an important parameter for controlling size of the nanoparticle.

The bio-mimetically synthesized AuNPs was purified. The particle size and shape was established changing the composition of the reaction mixture. Figure 3A-3C showed the TEM images of the gold nanoparticles synthesized using different Padina gymnospora leaf broth concentrations (A: 1%, B: 5%, and C: 10%) with 1 mM HAuCl₄ at 75°C.

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Figure 1: UV-Vis spectra of gold nanoparticles recorded after the reaction of 1 mM HAuCl₄ solution with 5% aqueous leaves extract of Padina gymnospora for 12 h.

Figure 2: Effect of Macroalgae Leaf Extract (MLE) concentration on gold nanoparticle synthesis under controlled physico-chemical conditions.

Figure 3: (A) TEM micrograph recorded from a drop-coated film of gold aqueous solution formed by the reaction of 1 mM HAuCl₄ with 1% Padina gymnospora leaf broth at pH 9.5, (B) at 5% leaf broth concentration (C) 10% % leaf broth concentration. The scale bar corresponds to 20 nm, (D) Gold nanoparticles size distribution histogram, (E) Selected area of electron diffraction pattern recorded from one of the gold nanoparticles. The diffraction rings have been indexed with reference to fcc gold.
their optical, electronic, magnetic and catalytic properties. EDS profile (Figure 4) showed a strong gold signal along with copper because of the copper grid base used for the analysis [22]. We propose that the main biomolecules (fucoxanthin or flavonoids) with an intense broad band absorbance at 3,439 cm\(^{-1}\) was responsible for the reduction of Aurochloric acid to AuNPs as confirmed from FTIR analysis, from our earlier report [23] and was also similar in case of plant-mediated silver nanoparticles [24]. Having confirmed from characterization, the AuNPs synthesized at 5% of macroalgae biomass with diameter ranging from 8 to 21 nm was opted for cell cytotoxicity assay.

Having established cultured characteristics on growth for the reporter cell lines, the cells were treated with bioconjugated Au nanoparticles for up to 48 h to observe the effect of toxicity. Given the strong reaction of cells to Au nanoparticles illustrated in Figure 5, this incubation time should reliably detect even low or slow-acting toxicity and vice versa. We performed toxicity tests during both the logarithmic and stationary phases of cell growth following the different concentration of the Au nanoparticles. The Au clusters proved highly toxic in this assay. Notably, at higher AuNPs concentration there was an asymmetric accumulation of AuNPs in the periphery outside the cell nucleus of the HepG2 cells. The results clearly indicated the aggregation of particles inside the cell cytoplasm of liver cancer cell with an IC50 value of about 82.91 nM induced 50% reduction in cell viability in comparison to lung cancer cell at 144.16 nM concentration (Figure 5). Figure 6A and B demonstrates the cellular morphology change in the HepG2 cell line along with the A549 cell line. The phase contrast microscopic images clearly showed that the morphology changes in A549 cell line were less significant. The extent of roundedness of the cell; one of the characteristics of stressed cells was measured upon treatment with gold nanoparticles.

Taken together these results suggest a stringent size dependency of the cytotoxicity of nanoscopic gold clusters. Thus the synthesized nanoparticles were found to be potently cytotoxic agent against HepG2 cell lines and mildly cytotoxic against A549 cell lines. However, further study is needed to understand the exact mechanism of anticancer activity of bio-functional gold nanoparticle. Algal pigments, such as fucoxanthins, a kind of carotenoids rich in hydroxyl groups very abundant in polysaccharides of the algal cell wall could have been adsorbed on the surface can stimulate or suppress the immune system due to the presence of −OH groups. Presence of such phenolic moieties may be assumed to have synergic effect for the anti-proliferative activities of these bio-adsorbed metal nanoparticles [25]. The other reason of anti-proliferative activity may be due to anisotropic nanoparticle used in this study [26]. These shape-dependent properties of AuNPs have different behaviour and make them suitable for therapeutic utilization [27,28]. AuNPs of certain non-regular shapes can readly be adsorbed to the surfaces of the biomolecules which show higher surface plasmon resonance and will have a greater contrast effect than those of photothermal dyes that are used regularly in detection of cancer cells [29].

**DNA fragmentation**

The cellular metabolic activity seemed affected by the AuNPs, thus, the possibility of apoptosis induction by the nanoparticles was assessed.
DNA fragmentation was performed for the tumour cells treated with purified gold nanoparticle and it was determined by visualization of DNA laddering on agarose gel electrophoresis. DNA Fragmentation of the HepG2 cells treated with different concentrations such as 79, 80, 81, 82 and 83nM (Lane L3, L4, L5, L6, and L7) was performed. DNA ladders of the corresponding treated samples confirmed apoptosis (Figure 7A) and showed that the AuNPs-treated Hep G2 cells exhibited extensive double strand breaks, thereby yielding a higher ladder appearance (Lane 6), while the DNA of control HepG2 cells exhibited minimum breakage (Lane 2). The 1 kb ladder (Lane 1) was used to find the molecular weight of cleaved DNA fragments.

DNA Fragmentation of the A549 cells treated with different concentrations such as 140, 141, 142, 143 and 144nM in Lanes L3, L4, L5, L6 and L7 respectively was performed (Figure 7B), and Lane L3 to L5 showed weak ladder pattern of DNA. DNA ladders of the corresponding treated samples confirmed apoptosis. Fragmented DNA showed a ladder-like pattern of Lane L1 that was the molecular marker while lane L2 was the negative control (untreated cells). Lane L7 showed high ladder pattern of DNA in A549 cells at concentration of 144nM. These changes in DNA represent the biochemical changes involved in the fragmentation of chromosomes into nucleosome units [30]. Fragmented DNA during apoptosis appears as a series of bands, which are described as “DNA ladders" on agarose gels, representing formation of oligonucleosomes with the characteristics of apoptosis [31]. The results of the present study showed, oligonucleosomal DNA laddering in the HepG2 and A549 tumour cells. The laddering pattern resulted in HepG2 and A549 tumour cells producing high molecular weight DNA fragments. This production of megabase-sized DNA fragments is reported to be associated with the detachment of cells from the monolayer and decreased cell volume that does not disturb membrane integrity [32].

**Morphological changes of apoptosis**

Biofunctionalized bioconjugate gold nanoparticles treatment on HepG2 and A549 cells were stained with Annexin-V and morphological changes were immediately observed under fluorescence microscopy. HepG2 cell treated with 82.91nM and A549 cell treated with 144.16nM of gold nanoparticles, showed an increase of propidium iodide positive dead cell (Red colour) in both, Annexin-V FITC positive cell (green colour) (Figure 8A), and cells which are both propidium iodide and Annexin VFITC positive (scattered greens orange); this indicates that treatment of gold nanoparticle in HepG2 cells (Figure 8B), positive A549 cells (Fig. 8C) and treated A549 cancer cells (Figure 8D) caused apoptosis. The staining confirmed the efficiency of biofunctionalized gold nanoparticles causing apoptosis. Investigation of apoptosis via Annexin V staining has also been reported, e.g., for TiO2 nanoparticles [33] and fullerences [34]. The change in nuclear morphology was determined by staining with the DNA-binding Hoechst dye in the presence of the biofunctionalized gold nanoparticle. Gold nanoparticle treated HepG2 and A549 cells were stained by Hoechst and morphological changes were immediately observed under a confocal fluorescence microscope (CFM). Untreated control cells exhibited blue-nuclei (Figure 9A), whereas, HepG2 cell treated with 82.91nM (Figure 9B) and A549 cell treated with 144.16nM of gold nanoparticles exhibited blue- nucleus with fluorescence, showing condensed chromatin in bright blue areas when compared to control cells (Figure 9C and 9D). Gold nanoparticle induced apoptotic morphological changes on HepG2 and A549 cells were observed after 24h treatment. The cells were then stained by acridine orange (AO) and the change in morphology were immediately observed using confocal microscopy. Untreated control cells exhibited bright green-nuclei fluorescence (Figure 10A), whereas, HepG2 cell treated with 82.91 nM (Figure 10B) and A549 cell treated with 144.16nM of gold nanoparticle exhibited orange- nucleus fluorescence (Figure 10C) when compared to control lung cancer cell (Figure 10D). It showed condensation of chromatin and DNA fragmentation as dense orange areas; this dense orange area indicated mitochondrial depolarization. The irregular size was due to loss of cellular volume associated with cytoskeletal breakdown and plasma membrane blebbing. These changes are the hallmark of cell undergoing apoptosis [35]. Assessments of membrane integrity with vital dye, such as trypan blue are nonspecific to determine the mode of cell death such as apoptotic cells unless morphological characteristics can be analyzed with the dyes [36]. Therefore, here Acridine orange (AO) a cell-permeable DNA-binding dye was used to analyze the morphological changes that occurred in the DNA. The results of this staining revealed that the gold nanoparticle induced the nuclear condensation and cell fragmentation.

**Conclusion**

The present study explores the potential antitumor activity of biologically synthesized AuNPs in a lung and liver tumor system in vitro by activation of molecular mechanism of cell death. The drug delivery systems are mainly developed according to their ability to differentiate...
Figure 9: CFM - Hoechst staining: (A) Control HepG2 cell, (B) Gold nanoparticle induced chromatin condensation in HepG2 cancer cell (C) Control A549 cell, (D) Gold nanoparticle induced chromatin condensation in A549 cancer cells.

Figure 10: CFM - Acridine orange staining: (A) Control HepG2 cell (B) Gold nanoparticle induced chromatin condensation and DNA breakage in HepG2 cancer cell and (C) Control A549 cancer cells (D) Gold nanoparticle induced chromatin condensation and DNA breakage in A549 cancer cells.
between malignant and non-malignant cells, making them a promising alternative to existing drugs. This type of targeting efficiency of AuNPs can be accomplished in future therapies using peptide conjugation, which directly targets tumor cells without affecting normal cells. Thus, a study of the exact mechanism by which AuNPs inhibit signaling cascades responsible for the development and progression of the disease would be a tremendous breakthrough in the field of nanomedicine and make these agents an effective alternative in tumor and angiogenesis-related diseases. Overall results obtained during this work have shown that the targeted gold nanoparticles have shown promising results in in vitro studies and acted as a drug carrier. This exploration creates the new avenue to a new standard where the different carotenoids and alkaloids-functionalized gold nanoparticles can be a powerful weapon in the field of nanomedicine. Thus, the development of biosynthetic approach and biofunctionalized targeted gold nanoparticles as therapeutic agents will generate great interest in both academy and industry.

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