Growth laws and self-similar growth regimes of coarsening two-dimensional foams: Transition from dry to wet limits

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(Dated: May 5, 2014)

We study the topology and geometry of two dimensional coarsening foams with arbitrary liquid fraction. To interpolate between the dry limit described by von Neumann’s law, and the wet limit described by Marqusee equation, the relevant bubble characteristics are the Plateau border radius and a new variable, the effective number of sides. We propose an equation for the individual bubble growth rate as the weighted sum of the growth through bubble-bubble interfaces and through bubble-Plateau borders interfaces. The resulting prediction is successfully tested, without adjustable parameter, using extensive bidimensional Potts model simulations. Simulations also show that a self-similar growth regime is observed at any liquid fraction and determine how the average size growth exponent, side number distribution and relative size distribution interpolate between the extreme limits. Applications include concentrated emulsions, grains in polycrystals and other domains with coarsening driven by curvature.

Liquid foams, namely gas bubbles separated by a continuous liquid phase, are ubiquitous [1, 2]. In floating foams as beer heads, ocean froths or pollutant foams, the fraction $\phi$ of their volume occupied by the liquid decreases with height, varying from a dry foam at the top to a bubbly liquid at the foam-liquid interface.

Since pressure can differ from one bubble to another, gas slowly diffuses. Some bubbles disappear and, as no new one is created, the average size increases. Foam coarsening is analogous to that of concentrated emulsions, grains in polycrystals, or two-phase domains where interface dynamics is driven by curvature. Its dynamics depends mainly on $\phi$, up to a material-specific time scale determined by the foam physico-chemistry [1, 2].

Understanding foam coarsening requires two different levels. First, the individual bubble growth law, which rules a bubble’s growth rate according to its size or shape. This law can be stated as a static geometry problem, and may be obtained analytically or by detailed bubble shape simulations. Second, the effect of such individual growth on the statistics of the foam, i.e., bubble size and topology distributions, requires statistical theories or large bubble number simulations.

In the very dry limit $\phi \to 0$, bubbles are polyhedra with thin curved faces meeting by three along thin lines called Plateau borders. Coarsening in that limit has been investigated experimentally, numerically and theoretically in two (2D) [3,7] and later in three dimensions (3D) [8,14].

In the very wet limit $\phi \to 1$, bubbles are round, dispersed in the liquid and far from each other, forming a “bubbly liquid” rather than a foam stricto sensu. Their coarsening follows “Ostwald-Lifschitz-Slyozov-Wagner” ripening, in 3D [15] [16] and later in 2D [17] [18].

In both limits, the foam eventually reaches a self-similar growth regime: statistical distributions of face numbers and relative sizes become invariant. Only the average size $\langle R \rangle$ grows in time, as a power law $\langle R \rangle \sim t^\beta$, with $\beta = 1/2$ in the dry limit and $\beta = 3/3$ in the wet one, reflecting that the underlying physical processes are different. The number $N$ of bubbles thus decreases as $t^{-2\beta}$ in 2D and $t^{-3\beta}$ in 3D. The growth law for intermediate liquid fractions has been addressed in experiments [19] and simulations [20, 21], but still lacks a unified theoretical description.

Here we address the 2D case. We propose a growth law to interpolate for $0 < \phi < 1$, with two parameters (diffusion coefficients) which are determined in each limit.

To test our prediction on 2D foam coarsening experiments is difficult, because we are not aware of any study where $\phi$ is systematically varied and precisely measured, or even rigorously defined. We rather use numerical simulations based on Potts model, suitable for large bubble numbers [10] [22] [24]. Beside testing our prediction, simulations also show that, for any $\phi$, side number and relative size distributions reach a self-similar growth regime where $\langle R \rangle$ grows as $t^\beta$. Values of $\beta$ interpolate between 1/2 and 1/3.

In the 2D dry limit, gas diffuses through neighbor bubbles walls, due to the pressure difference between bubbles related with wall curvature. The walls are curved because they meet at threefold vertices, forming equal angles of $2\pi/3$. Each vertex is responsible for a turn of $\pi/3$ in the vector tangent to the bubble perimeter. Consequently bubbles with 5 sides or less are convex, while bubbles
with 7 sides or more are concave, so that the walls curvature plus a turn of $\pi/3$ at each vertex sums up to $2\pi$. The resulting growth dynamics is von Neumann’s law [3]:

$$\frac{da^i}{dt} = -D_d \left(2\pi - n^i \frac{\pi}{3}\right) = \frac{\pi D_d}{3} (n^i - 6) , \quad (1)$$

where $a^i$ and $n^i$ are, respectively, area and number of sides of the $i$th bubble; $t$ is time; $D_d$ depends on the foam composition and is expressed in $\text{m}^2\text{s}^{-1}$ as a diffusion coefficient. Remarkably, the rhs of eq. (1) involves only the bubble’s number of sides and not its size or shape. At any time, bubbles with $n^i < 6$ shrink while bubbles with $n^i > 6$ grow. Since for topological reasons the average bubble number of sides is 6 [11], eq. (1) is compatible with gas volume conservation in the whole foam.

In the 2D wet limit, gas bubbles are dispersed in a liquid matrix. Dynamics is a consequence of pressure difference in the gas contained in a bubble or dissolved in the liquid. This pressure difference is proportional to the wall curvature, which for a circular bubble is the inverse of its radius, $R$. Marqusee [17] wrote the growth law using only $R$:

$$\frac{da^i}{dt} = 2\pi R^i \frac{dR^i}{dt} = 2\pi D_w f(R^i), \quad (2)$$

where $D_w$ is another diffusion coefficient-like constant as above, and

$$f(R^i) = \frac{R^i}{\xi} \frac{K_i \left(\frac{R^i}{\xi}\right)}{K_0 \left(\frac{R^i}{\xi}\right)} \left[\frac{1}{R_c} - \frac{1}{R^i}\right], \quad (3)$$

where $K_j$s are $j$th order modified Bessel functions of second kind; $\xi$ is the screening length (roughly, the typical distance beyond which bubbles do not feel the influence of each other); and $R_c$ is the critical radius for which there is no growth, calculated by imposing total gas volume conservation, i.e. $\frac{d}{dt} \sum a^i = 0$. At any given time, bubbles with radius smaller than $R_c$ lose gas while those with radius larger than $R_c$ gain gas.

Interpolating between eqs. (1) and (2) seems difficult because they use very different variables: number of sides $n^i$ and radius $R^i$. However, both equations involve the product of curvature times length (i.e. angle, Fig. 1) of the interfaces through which the gas diffuses. We propose that for any $\phi$ the bubble growth rate is simply the superposition of growth through the interfaces shared either directly with other bubbles or with Plateau borders. It can thus be calculated as the weighted average of eqs. (1) and (2). The weights are fractions of $2\pi$ angle carried by dry or wet parts of the $i$th bubble perimeter, $\Theta_d^i$ or $\Theta_w^i$, which are sums of angles $\theta_d$ or $\theta_w$ carried, respectively, by all dry or wet interfaces of the bubble (Fig. 1), such that $\Theta_d^i + \Theta_w^i = 2\pi$. We now detail how to perform this linear superposition, which turns out to be unexpectedly successful at all $\phi$.

In dry foams, at each vertex $\theta_d = \pi/3$, so that $\Theta_d^i = n^i\pi/3$. In wet foams, $\Theta_w = 2\pi$. To interpolate, we characterize the bubble $i$ by introducing its effective number of sides defined for any $\phi$ as:

$$n_{eff}^i = \frac{3}{\pi} \Theta_w^i - 6 = \frac{3}{\pi} \Theta_d^i . \quad (4)$$

Although $n_{eff}^i$ conveys no more information than $\Theta_d$ or $\Theta_w$, its value is intuitive: $n_{eff}^i = n^i$ for a polygonal dry bubble; $n_{eff}^i = 6$ for a circular wet bubble; $1 - n_{eff}^i/6$ or $n_{eff}^i/6$ are exactly the fractions of $2\pi$ carried by either dry or wet interfaces, respectively. Similarly, we characterize the bubble $i$ by the curvature radius $R_w^i$ of its Plateau borders; $R_w^i = 0$ for a dry bubble and $R_w^i = R^i$ for a wet bubble. With these variables $n_{eff}^i$, $R_w^i$, we can weight eqs. (1) and (2) and interpolate for any $\phi$:

$$\frac{da^i}{dt} = \frac{\pi}{3} \left[D_d(n_{eff}^i - 6) + D_w n_{eff}^i f(R_w^i)\right] . \quad (5)$$

To test eq. (5) we have implemented extensive Potts model simulations in the spirit of [10]. Like in experimental pictures, a simulation represents gas bubbles and liquid phase as connected regions on a square lattice of pixels. To ensure that liquid is evenly distributed between Plateau borders (fig. 1), and that $\phi$ is conserved, we represent the liquid phase as a fixed number of tiny “liquid drops” of fixed area, and each Plateau border contains several of these drops. Each interface between gas bubbles represents a thin film made of two gas-liquid interfaces which mildly repel each other. We then assign to the interfacial energy between gas bubbles a value $(2-\varepsilon)$ times the interfacial energy between gas bubble and liquid drop. Experimentally, in soap foams $\varepsilon$ is typically of order $10^{-3}$. Its precise value is not crucial in what follows, as long as $\varepsilon \ll 1$ (and $\varepsilon > 0$ to avoid numerical instabilities). Here we choose $\varepsilon = 0.01$ as a compromise between realism and computing speed. Since liquid drops are free to cluster together, we assign to drop-drop interfaces an energy $10^3$ times smaller. The liquid is sucked in the Plateau borders (for reviews see [11, 23]).

There are $N = 8944^2$ sites; each site $s$ is assigned with a label, $S(s)$. There are $N$ bubbles ($S = 1$ to $N$) and
$N_w$ drops ($S = N + 1$ to $N + N_w$), assigned with a type $\tau = 1$ or 0, respectively. A configuration has an energy:
\[
E = \sum_{s=1}^{N} \sum_{s'=1}^{36} J(\tau, \tau') [1 - \delta_{s, s'}] + \lambda \sum_{s=1}^{N+N_w} \left[ a(S) - a_t \right]^2,
\]
where $s'$ stands for the sum over the first 36 neighbors of the site $s$, to avoid pinning to the grid \cite{27} and to extend the range of the disjoining pressure up to three pixels; $S$ and $S'$ are the labels of sites $s$ and $s'$, respectively; $\delta$ is the Kronecker symbol; $\tau$ and $\tau'$ are the types of $S$ and $S'$, respectively; $J(1, 1) = 1.99$ $J(1, 0)$, $J(1, 0) = J(0, 1) = 0.7$, and $J(0, 0) = 0.001$ are the interfacial energies; $a(S)$ is the current area of liquid drop $S$; $a_t = 8$ pixels is a target area common to all drops; $\lambda = 9$ penalises any deviation from $a_t$.

Simulations begin with $N_0 = 2 \times 10^5$ gas bubbles randomly dispersed over the grid, with smooth interfaces and a normal distribution of areas around the average $\langle N \rangle/N_0$. For $\phi > 0$, in the initial configuration each vertex contains at least one drop. The total number of drops and the initial average area of gas bubbles are set according to the desired $\phi$.

The simulation dynamics follows Monte Carlo method. We randomly choose a site, temporarily change its label to the value of one of its neighbors, and calculate the change in energy $\Delta E$. If $\Delta E \leq 0$ this relabeling is accepted. If $\Delta E > 0$ the change is accepted with probability $\exp(-\Delta E/T)$, where $T$ is the fluctuation allowance, here taken as $T = 3$ to escape possible metastable states.

We measure diffusion coefficients as follows. We first perform a simulation in the dry limit $\phi = 0$. Plotting the rhs and lhs of eq. \cite{1} determines by linear regression $D_d = 1.68$. We then perform a simulation in the wet limit $\phi = 0.9$. Plotting the rhs and lhs of eq. \cite{2} determines by linear regression $D_w = 0.78$. With these parameters, together with area and wet and dry perimeters for each bubble from simulations, eq. \cite{5} predicts without any adjustable parameter each bubble growth for any intermediary $\phi$.

We measure angles on a square grid as follows. The simulation is halted. We add over all dry interfacial pixels of bubble $i$ the probability that it would grow (or shrink) over (yielding to) other bubbles, calculated by the Monte Carlo method. This determines the growth rate of bubble $i$ through its dry interfaces, $G_d$, and thus $\Theta^i_d = D_d G^i_d$.

We then obtain $\Theta^i_w = 2\pi - \Theta^i_d$, and $R_w^i = \frac{P_i}{\Theta_w}$, where $P_w^i$ is the wet interface length. During this measure, no change is performed. The simulation then resumes.

We run simulations with $\phi = 0$, $0.02$, $0.06$, $0.18$, $0.36$, $0.54$, $0.72$ and $0.90$. For any $\phi$, the evolution of the gas bubble number has a power law behavior (Fig. 2(a)), which is compatible with self-similar growth regimes. Fig. 2(b,c) shows that the power law exponent $\beta$ varies as $\beta(\phi) \approx 1/2 - \phi^{0.2}$. Thus $\beta(\phi)$ decreases continuously from $1/2$ to $1/3$, the expected limit values, with $d\beta/d\phi$ diverging at $\phi \to 0$. It would be interesting to explain theoretically this variation of $\beta$ with $\phi$.

Fig. 3 shows snapshots for different $\phi$s after 20,000 Monte Carlo Steps (MCS). The liquid accumulates at the vertices for small $\phi$s, and, as it increases, liquid goes also between bubbles. The ratio $d/\xi$, where $d$ is the typical distance between bubbles, estimated as $d = 2(\sqrt{N/(N\pi)} - \sqrt{\langle a \rangle}/\pi)$, and $\xi$ the screening length (eq. 2), increases with $\phi$ (Fig. S1 of \cite{28}). At $\phi = 0.54$, $d/\xi > 1$, and at $\phi = 0.90$ the bubbles are not touching each other.

Fig. 4 presents the average area growth rate of gas bubbles versus $R_w$, and in insets the distribution function of $R_w/(R_w)$. The superposition of plots taken at different times indicates a self-similar growth regime. The agreement with the theoretical prediction (eq. 5) is excellent. For $\phi = 0$ all bubbles have $R_w = 0$ and this plot does not convey any information. For $\phi = 0.02$ and 0.06 the noise arises from measuring the curvature radius of Plateau borders. For different values of $R_w$, distributions of bubble growth rates are presented in Fig. S2 of \cite{28}.

Fig. 5 presents the average area growth rate of gas
bubbles versus the effective number of sides, $n_{eff}$. Again, the agreement with the theoretical prediction (eq. 5) is excellent. For $\phi = 0.90$ all bubbles have $n_{eff} = 6$ and this plot does not convey any information. For different values of $n_{eff}$, distribution functions of bubble growth rates are plotted in Fig. S3 of [28]. Plots of growth rates versus $n_{eff}/\langle n_{eff} \rangle$ (Fig. S4 of [28]) and versus $R_w/\langle R_w \rangle$ (Fig. S5 of [28]) discriminate the relative contributions of dry or wet interfaces to the growth.

This work has been partially supported by Brazilian agencies CNPq, CAPES, and FAPERGS, and initiated during visits of RdA and GLT to FG at the LSP/LIPhy, University of Grenoble.

\begin{thebibliography}{99}

[1] I. Cantat, S. Cohen-Addad, F. Elias, F. Graner, R. Höhler, O. Pitois, F. Rouyer, and A. Saint-Jalmes, *Les mousse: structure et dynamique* (Belin, Paris, 2010).
[2] D. Weaire and S. Hutzler, *Physics of Foams* (Oxford University Press, Oxford, 2001).
[3] J. Stavans, Rep. Progr. Phys 56, 733 (1993).
[4] J. von Neumann, in *Metal Interfaces*, edited by R. Brick, (ASM, Cleveland, OH, 1952) p. 108.
[5] J. Stavans, and J.A. Glazier, Phys. Rev. Lett. 62, 1318 (1989).
[6] V. Pignol, PhD Thesis, unpublished.
[7] J.R. Iglesias, and R.M.C. de Almeida, Phys. Rev. A 43, 2763 (1991).
[8] F. Wakai, N. Enomoto, and H. Ogawa, Acta Mater. 48, 1297 (2000).
[9] C.E. Krill III and L.-Q. Chen, Acta Mater. 50, 3059 (2002).
[10] G.L. Thomas, R.M.C. de Almeida, and F. Graner, Phys. Rev. E 74, 021407 (2006).
[11] J. Lambert, R. Mokso, I. Cantat, P. Cloetens, J.A. Glazier, F. Graner, and R. Delannay, Phys. Rev. Lett. 104, 248304 (2010).
[12] P. Streitenberger, and D. Zölner, Scr. Mater. 55, 461 (2006).
[13] S. Hilgenfeldt, A.M. Kraynik, D.A. Reimelt, and J.M. Sullivan, Europhys. Lett. 67, 484 (2004).
[14] R. Mc Pherson, and D. Srolovitz, Nature 446, 1053 (2007).
[15] W. Ostwald “Principles of Inorganic Chemistry,” (Macmillan, London, 1902) p. 58; “Grundriss der Allgem. Chemie” (Macmillan, London, 1908) p. 96; “Foundations of Analytic Chemistry” (Macmillan, London, 1908) p. 22, 3rd ed.
[16] I.M. Lifschitz and V.V. Slyozov, Zh. Eksp. Teor. Fiz. 35, 479 (1958) [Sov. Phys. JETP 8, 331 (1959)]; C. Wagner, Z. Elektrochem. 65, 581 (1961).
[17] J. Marqusee J. Chem. Phys. 80, 563 (1984).
[18] J.H. Yao, K.R. Elder, H. Guo, and M. Grant, Phys. Rev. B 45, 8173 (1992).
[19] J. Lambert, I. Cantat, R. Delannay, R. Mokso, P. Cloetens, J.A. Glazier, and F. Graner, Phys. Rev. Lett. 99, 058304 (2007).
[20] F. Bolton and D. Weaire, Philos. Mag. B 65, 473 (1992).
[21] S. Hutzler and D. Weaire, Phil. Mag. B 71, 277 (1995).
[22] M.P. Anderson, G.S. Grest, and D.J. Srolovitz, Phil. Mag. B59, 293 (1989).
[23] J.A. Glazier, M.P. Anderson, and G.S. Grest, Phil. Mag. A 62, 615 (1990).
[24] D. Zölner and P. Streitenberger, Scr. Mater. 54, 1697 (2006).
[25] V. Bergeron, J. Phys. Cond. Matter 11, R215 (1999).
[26] W.C. Graustein, Ann. of Math. 32, 149 (1931).
[27] E.A. Holm, J.A. Glazier, D.J. Srolovitz, and G.S. Grest, Phys. Rev. A 43, 2662 (1991).
[28] See EPAPS Document No.[ ] for additional plots.

\end{thebibliography}
FIG. 1: Ratio of average distance between air bubbles and screening length $\xi$ for different liquid fractions during scaling regime. Observe that, for $\phi \geq 0.54$, $d > \xi$. The red line is a linear fit.
FIG. 2: Normalized bubble growth probability distributions, for the liquid fractions $\phi$ presented in the paper (0.0, 0.02, 0.06, 0.18, 0.54, and 0.9), in the three distinct ranges of $R_w/\langle R_w \rangle$ shown in the upper left corner. Notice that, for $\phi = 0.0$, $\langle R_w \rangle = 0$. 
FIG. 3: Normalized bubble growth probability distributions, for the liquid fractions $\phi$ presented in the paper (0.0, 0.02, 0.06, 0.18, 0.54, and 0.9), in the ranges of $n_{\text{eff}}/\langle n_{\text{eff}} \rangle$ presented in the lower right corner. Notice that, as $\phi$ increases, the distributions narrow around $n_{\text{eff}}$, so no bubbles are found far from this $n_{\text{eff}}$ value. For $\phi = 0.54$ and $\phi = 0.9$ there are only bubbles with $n_{\text{eff}} \approx \langle n_{\text{eff}} \rangle$. 
FIG. 4: Area growth rate of air bubbles through the dry interfaces as a function of $n_{eff}/\langle n_{eff} \rangle$ for different liquid fractions and at different instants of the scaling regime. The insets present the probability density of $n_{eff}/\langle n_{eff} \rangle$ at the same instants.
FIG. 5: Area growth rate of air bubbles through the wet interfaces as a function of $R_w/\langle R_w \rangle$ for different liquid fractions and at different instants of the scaling regime. The insets present the probability density of $R_w/\langle R_w \rangle$ at the same instants.