Reply to Liao and Wynne: The size of crystal nucleus remains an open question

Takuji B. M. Adachi,1* Johanna Brazarda, and Oscar Urquidi

The letter by Liao and Wynne (1) lacks a description of key parameters used in their “back-of-the-envelope calculation.” It is hard to comment whether they simply made an error in calculation or the model may not be suited for this case (e.g., dimensionality of the heat equation) to obtain a 163-K increase. Liao and Wynne used Raman thermometry to support their claim and estimated an even higher temperature increase (400 K). Neither raw Raman spectra nor the wavenumber used for calculating temperature are shown, which makes it hard to point out what they did wrong. Raman thermometry requires careful measurement and analysis (5). Also, as a basic principle, one should avoid extrapolating data for 1.2 W using data from 10 to 50 mW.

Fig. 1 shows Raman spectra of SnCl₂ aqueous solution measured on our setup. The laser power was varied up to 1.2 W. If the temperature were elevated by the 532-nm laser, the relative peak intensity of Stokes Raman (I₆) to anti-Stokes Raman (I₈) would have decreased. (B) The expected ratio of I₆ to I₈ at 111 cm⁻¹ peak as a function of temperature calculated using the model in ref. 5. As expected, the I₆/I₈ decreases as temperature increases. (C) The I₆/I₈ calculated from the data shown in A. The ratio remains constant around room temperature even at 1.2-W laser power.

Author affiliations: 1Department of Physical Chemistry, Sciences II, University of Geneva, 1211 Geneva, Switzerland

Author contributions: T.B.M.A., J.B., and O.U. wrote the paper; and T.B.M.A., J.B., and O.U. performed the experiments and analysis shown in this letter.

The authors declare no competing interest.

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1To whom correspondence may be addressed. Email: Takuji.Adachi@unige.ch.

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Fig. 2. Demonstration of optical trapping of nanoparticles. (A) Fluorescence time trace of 120-nm polystyrene nanoparticles doped by fluorescent label (Spherotech Inc., light yellow) under 785-nm continuous-wave (CW) laser irradiation. The 785-nm laser works as an optical trapping beam and two-photon excitation beam. The number of trapped nanoparticles increases, and fluorescence intensity increases. Fluorescence microscope images also confirm that the laser spot becomes brighter due to the increased number of trapped nanoparticles. (Scale: 10 μm.) The arrows indicate the focused laser spot. Similar experiments were originally reported in ref. 8. (B) Bright-field microscope images of 60-nm polystyrene nanoparticles (Spherotech Inc.) under optical trapping by a 532-nm CW laser. (Scale: 10 μm.) The arrows indicate the laser spot. The size of particles is smaller than the diffraction limit of optical microscope; therefore, images appear blurry. The blurry objects at the center, however, gradually become more evident over the course of trapping as the number of trapped nanoparticles increases. Similar experiments were originally reported in ref. 9.

1.2 W. This sample shows distinct peaks and is better suited than the glycine/D2O solution used by Liao and Wynne. No temperature increase was observed. Moreover, movie S1 in ref. 2 shows no boiling.

The optical trapping model in the Rayleigh regime (6, 7) predicts that nanoscale objects can be stably trapped. There are many papers showing this. We reproduced the results reported in refs. 8 and 9 (Fig. 2) to reconfirm this fact. Liao and Wynne cite ref. 10 to claim that particles of <1 μm cannot be trapped. However, this reference discusses the trapping of nanomaterials. In Urquidi et al. (2), the glycine solution was filtered (200-nm pore) to avoid microscale dust particles, and no microscale particle is seen in movie S1 in ref. 2.

The title of the letter by Liao and Wynne (1) indicates that they wrongly interpreted our paper. We have never mentioned that oligomers are the cause of nucleation. We specifically wrote a sentence stating that, “in this work, we could not estimate the size of aggregates prior to nucleation.” Our contribution is proposing that the structure of glycine prenucleation aggregates may be hydrogen-bonded linear networks, which goes beyond the general description of prenucleation aggregates as “amorphous.” The size of the nucleus remains an open question. Our results show, however, that it is at least smaller than the diffraction limit of optical microscopy.

Liao and Wynne recently showed that laser can transform microparticles into crystals (11). It is, however, an overstatement to claim that all the other works in the field of laser-induced crystallization occur through the same path. Liao and Wynne (11) present no evidence to impose such a strong conclusion on the previous works from other experts.

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