ELICITATION OF THERAPEUTIC POTENTIAL AND OXIDATIVE STRESS ASSESSMENT OF FENUGREEK SPROUTS UNDER UV IRRADIATION

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INTRODUCTION

Since recent past years, seed sprouting technique is gaining more significance commercially all over the world. The sprouts being the initial phase of the growth and developmental process, their nutrient content is very high. Sprouts are a resource of the nutrients besides factors such as enzyme inhibitors from the seed that make sprouts safe for consumption. Sprouting in fenugreek is known to improve flavour, colour and taste of food. Fenugreek seed extract has antioxidant; anti-diabetic, anti-catarrh, antitoxic and immune-modulatory activities [8].

Free radicals are released due to an imbalance in pro-oxidant/antioxidant homeostasis and also during cellular metabolism. These free radicals and reactive oxygen species (ROS) being very unstable, react with the biomolecules leading to inactivation of metabolic enzymes and damaging important cellular components, consequently causing several chronic diseases such as Alzheimer’s disease, Parkinson’s disease, atherosclerosis, pancreatic diseases, cancer, ageing and other neurological [2, 3].

These toxic effects of ROS and unstable free radicals can be mitigated by the counter action of some bioactive compounds such as polyphenols, ascorbic acid, carotene, glutathione etc. Such antioxidant defences are extremely important as they represent the direct removal of free radicals, thus providing maximal protection for the biological system from these chronic and degenerative diseases [4, 5]. Hyperglycemia, a disorder characterized by an abnormal excess of sugar level in the blood, has been found to be associated with the onset of type 2 diabetes mellitus and other cardiovascular complications [6, 7]. Therefore the natural sources of antioxidant and anti-hyperglycaemic compounds are very important for consumption for better health management.

Fenugreek, Trigonella foenum-graecum L. is used as a traditional medicine since ancient times. It is commonly known as seed spice used to improve flavour, colour and taste of food. Fenugreek seed extract has antioxidant; anti-diabetic, anti-catarrh, antitoxic and immune-modulatory activities [8].

Ultraviolet (UV) radiation despite of its deleterious impact on plant systems including deoxyribonucleic acid (DNA) damage, mechanistic damage to photosynthetic apparatus and degradation of protein synthesis [9] has been positively implemented in various researchers with sublethal doses. Several reports have suggested that UV radiations have been successfully used in the extension of shelf life of fruits and vegetables. The application of UV treatment in several plant systems has resulted in enhancement of bioactive phytochemical content and quality composition [10].

Existing literature have suggested that UV radiation has been found to enhance the synthesis of bioactive components in various plant samples but till date, no research has been reported on the effect of UV radiation on the phytochemical content, antioxidant and anti-diabetic property of fenugreek during post-germination phases. So the objective of this study was to evaluate the alteration in antioxidant activity, anti-diabetic property and also the associated phytochemical contents of fenugreek sprouts in different germination phases elicited by UV irradiation.

MATERIALS AND METHODS

Chemicals and reagents

DPPH, ABTS, methanol, ferrozine, FeCl₃, potassium ferricyanide, trichloroacetic acid, α-amylase, DNSA, starch, sodium carbonate, p-
nitrophenyl α-glucoside (pNPG), α-glucosidase enzyme, quercetin, AlCl₃, NaOH, NaN₃, gallic, folin-ciocalteau, FeCl₃, Evans blue, CaCl₂, potassium sulphate, Schiff’s reagent, nitroblue tetratolium, potassium iodide, starch, acrylamide, N,N'-methylene-bis (acrylamide), potassium hydroxide, potassium dihydrogen phosphate, sodium dihydrogen phosphate, EDTA, triton, riboflavin, TEMED, H₂O₂, benzenide, Tris, NADPH were purchased either from Sigma chemicals (USA) or Merck (India) of analytical grade.

Elicitation process

The seeds of Trigonella foenum-graecum (Methi-P. E. B) were purchased from National Seeds Corporation Limited, Beej Bhavan, Pusa Complex, New Delhi. The seeds were surface sterilized with 1% (v/v) sodium hypochlorite solution. The sterilized seeds were elicited with UV rays of different wavelength i.e., 254 nm for 10 min, 365 nm for 10 min and both 254 nm and 365 nm in combination (COMB) for 5 min each. The dose applied were equal to 1.27 kJ m⁻², 1.94 kJ m⁻² and 1.61 kJ m⁻² for 254 nm, 365 nm and combination respectively. Seeds were treated with UV radiation for 3 times a day with 8h of the interval during the course of germination up to 4 d. The untreated seeds were used as a control set. For UV exposure the fenugreek seeds were placed under a 6 watt germicidal UV lamp (VL-6, LC 254 nm, Viber Lournart, France) at a distance of 5 cm from seeds. The intensity of irradiation was determined using a photo-radiometer (HD2102.2, Delta OHM, Padova, Italy).

Preparation of extract

The sprouts of fenugreek of 4 different stages (24h, 48h, 72h and 96h) treated and untreated were crushed in mortar-pestle and processed through soxhlet extraction apparatus with methanol. The obtained extract was concentrated to a definite concentration of 1g/ml using a vacuum rotary evaporator at temperature ≤ 45 °C and stored in brown bottles for further experimental analysis.

DPPH based free radical scavenging activity

The radical scavenging activity of the extracts was measured by 2, 2-diphenyl-1-picyrylhydrazyl (DPPH) method [11]. The reaction mixture containing 1.8 ml of 0.1 mmol DPPH and 0.2 ml extracts were incubated for 30 min and the absorbance was recorded at 517 nm.

ABTS⁺ radical cation(·)s decolorization assay

The analysis of 2,2’-azinobis-(3-ethybenzthiazoline  sulfonic acid) (ABTS⁺) radical cation(·)s scavenging activity was determined by spectrophotometric method [12]. This method is based on the ability of antioxidants to quench the ABTS⁺ radical cation, a blue/green chromophore with characteristic absorption at 734 nm.

Metal chelating activity

The chelating activity of the extracts for ferrous ions Fe²⁺ was measured spectrophotometrically [13]. To 0.4 ml of methanol extract, 1.6 ml of methanol was diluted and mixed with 0.04 ml of FeCl₃ (2 mmol). After 30 sec, 0.8 ml ferrozine (5 mmol) was added. After 10 min at room temperature, the absorbance of the Fe²⁺-Ferrozine complex was measured at 562 nm.

Ferric reducing antioxidant power (FRAP) assay

The reducing antioxidant power of plant methanolic extracts was determined by the standard method [14]. Different concentrations of 1 ml of fruit extracts were mixed with phosphate buffer (2.5 ml, 0.2 M, pH 6.6) and potassium ferricyanide [K₃Fe(CN)₆] (2.5 ml, 1%). The mixture was incubated at 50 °C for 20 min. Then, 2.5 ml of trichloroacetic acid (10% w/v) was added to the mixture, which was then centrifuged for 10 min at 3000 rpm. The upper layer of solution (8.5 ml) was mixed with distilled water (2.5 ml) and 1% FeCl₃ (0.5 ml). The absorbance was measured at 700 nm against a blank using UV-VIS spectrophotometer (Sysstronics, 2201, India).

Total phenol content

Total phenol content of plant extracts was determined by folin-ciocalteau method [15]. A volume of 0.5 ml of the plant extract was mixed in a test tube containing 1 ml of 95% ethanol, 5 ml of distilled water and 0.5 ml of 50% folin-ciocalteau reagent. The resultant mixture was allowed to react for 5 min and 1 ml of 5% Na₂CO₃ was added. It was mixed thoroughly and placed in the dark for 1 h. Finally, the absorbance of the coloured reaction product was measured at 765 nm against the reagent blank. The total phenolic content was expressed as mg of gallic acid equivalent (GAE) per gram fresh weight (fw).

Total flavonoid content

The flavonoid content was measured following a standard spectrophotometric method [16]. One ml of methanolic extracts was diluted with water (4 ml) in a 10 ml volumetric flask. Initially, 5% NaNO₂ solution (0.3 ml) was added to each volumetric flask; at 5 min, 10% AlCl₃ (0.3 ml) was added; and then after 6 min, 1M NaOH (2 ml) was added. After addition of distilled water (24 ml) absorbance was read at 510 nm. The total flavonoid content in different extracts was calculated as quercetin equivalent (QE) per gm fresh weight.

α-glucosidase inhibitory activity

The α-glucosidase inhibitory activity was determined following the protocol suggested by Ghosal and Mandal [17]. A volume of 250 μl of plant extract was reacted with 500 μl of the α-glucosidase solution (0.1 U/ml in 0.1M potassium phosphate buffer, pH 6.9) in test tubes and incubated at 37 °C for about half an hour. After incubation, 250 μl of p-nitrophenyl α-glucoside (pNPG) (5 mMol in 0.1 M phosphate buffer, pH 6.9) was added to each test tube and the reaction mixture was kept for 30 min at 37 °C. To the above reaction mixture 300 μl of sodium carbonate solution (100 mMol) was added and incubated again for 20 min at 37 °C. The absorbance was recorded at 405 nm and compared to the control in which 250 μl of buffer was added instead of plant extract. The α-glucosidase inhibitory activity was evaluated by measuring the release of p-nitrophenol from pNPG.

α-amylase inhibitory activity

The α-amylase inhibitory potential of the sample was estimated by spectrophotometric method [17]. At first 500 μl of extract was reacted with 500 μl of α-amylase solution (0.5 mg/ml in 0.02 M sodium phosphate buffer, pH 6.9) and incubated at 37 °C for 5 min. After incubation, 500 μl of starch solution (1%, w/v) was added and further kept undisturbed for 10 min at room temperature. The reaction was terminated by addition of 1 ml of 3,5-dinitrosalicylic acid (DNSA) reagent and the test tubes were placed in a hot water bath till the colour of reaction mixture changed to orange-red. After that, the reaction mixture was cooled and diluted with distilled water up to 5 ml. The generation of maltose was determined by recording the OD value at 540 nm.

Histochemical detection

For detection of hydrogen peroxide localisation in the roots, the seedlings were kept submerged for about 45 min in potassium iodide (0.1M)/starch (4% w/v) reagent [18]. The histochemical detection of superoxide localisation, lipid peroxidation and plasma membrane integrity was conducted by the method followed by Kaur et al. [19]. For detection of superoxide anion radical the seedlings were submersed in 50 mMol phosphate buffer (pH 7.5) containing (0.05% w/v) nitroblue tetrazolium (NBT) for 20 min in presence of light and then the roots were placed in ethanol for terminating the reaction. For detection of lipid peroxidation, the fenugreek roots were placed in Schiff’s reagent until the coloured stain developed on the roots. The extra stain imparted by the reagent was removed using Evans blue solution (0.025% w/v in 100 mMol CaCl₂) for about half an hour and then the stained roots were rinsed thrice with distilled water for removal of extra stain. All the above-stained roots were photographed using Nikon SLR camera (Model D3200).
Detection of isozyme pattern by gel electrophoresis

The tissues were homogenized with ice-cold 50 mmol potassium phosphate buffer (pH 7.0) including 1% polyvinyl pyrrolidone, 2 mmol ethylene diamine tetraacetic acid (EDTA) and 0.1% triton. The homogenate was centrifuged at 10,000g for 20 min at 4 °C and the obtained supernatant was used for further detection of isoenzymes.

The isozymes of superoxide dismutase (SOD; EC 1.15.1.1), peroxidase (POD; EC 1.11.1.7) and NADPH oxidase (NOX; EC 1.6.3.1) were separated on discontinuous polyacrylamide gels (composed of 5% stacking gel and 10% resolving gel) and for catalase (CAT; EC 1.11.1.6) (4% stacking gel and 8% resolving gel) under non-denaturing conditions. Proteins were electrophoresed at 4 °C and 80V in the stacking gel followed by 120V in the resolving gel.

The isozyme pattern of superoxide dismutase activity was detected on gel as described by Pereira et al. [20]. The gels were rinsed in distilled water and incubated in an assay mixture containing 0.05M potassium phosphate buffer (pH 7.8), EDTA (1 mmol), riboflavin (0.05 mmol), nitroblue tetrazolium (0.1 mmol) and 0.3% N,N',N'-tetramethyl-ethylene diamine (TEMED) in the dark condition for 30 min at room temperature. After incubation, the gels were again rinsed with distilled water and then illuminated on a light box until the colourless bands of SOD appeared against a purple-stained gel background.

For detection of catalase isoforms, gels were soaked in 50 mmol potassium-phosphate buffer (pH 7.0) and then in the same buffer containing 5 mmol H2O2. After incubation, gels were rinsed with distilled water and stained in a reaction mixture composed of 1% potassium ferricyanide and 1% ferric chloride [21].

For peroxidase isoforms, the gels were soaked for 20 min in 0.2 M acetate buffer (pH 5.5) with 5 mmol benzidine and 5 mmol H2O2 [22]. For NOX isoforms, the gels were soaked in gel staining solution containing NBT (0.5 mg/ml) and 134 µM NADPH in 10 mmol Tris buffer, pH 7.4 [23].

Statistical analysis

Each experiment was performed in triplicate. The software package, MS Excel 2007 (Microsoft, Redmond, WA, USA) was used for comparing the antioxidant and antidiabetic attributes of different stages of the fenugreek sprouts. The different group means were compared by Duncan’s Multiple Range Test and differences at p<0.05 was considered significant in all cases.

RESULTS

Phytochemical quantification

For phytochemical analysis, total phenol and flavonol content was determined. The total phenol and flavonol content of fenugreek sprouts was estimated for 4 d of germination. The highest stimulation was observed at the 48h stage for control as well as those subjected to UV exposure. Among the different wavelength of UV light applied, the sprouts treated with UV 365 nm exhibited the highest phenolic content which was 15% higher than control (fig. 1A). From the estimation of flavonol content, a gradual increase in the flavonol content was observed towards germination time. The maximum flavonol stimulation was observed for the sprouts treated with UV 254 nm which was about 50% greater than control at 72h stage; UV 365 nm was found to enhance the highest flavonol content at 48h (41%) and 96h (34%) and combined exposure enhanced 56% of flavonol content at 24h of germination (fig. 1B).

![Fig. 1: A) Total phenol content and B) total flavonol content of the fenugreek sprouts of 24h, 48h, 72h and 96h stage, untreated and treated with UV radiation of wavelength 254 nm, 365 nm and combined (254 nm+365 nm). Results are represented as mean±SEM, n=3. Values with different letters (a, b, c etc) are significantly (P<0.05) different from each other by Duncan’s multiple range test (DMRT)](Image 100x275 to 513x446)

Free radical scavenging activity

DPPH and ABTS assay is widely used to evaluate the free radical scavenging activity of different plant materials. In the case of DPPH assay, extract having lowest IC50 value possessed higher scavenging activity. In the present experiment 48h seedlings exhibited the best activity compared to other germination stages and the seedlings treated with UV 254 nm showed the lowest IC50 value (34.73 mg ml−1) followed by UV 365 nm (47.95 mg ml−1) and combination of both (65.79 mg ml−1) also showed better activity than control set (77.9 mg ml−1). Free radical scavenging activity of seedlings for DPPH has increased up to two-fold in treatment with UV 254 nm (34.73 mg ml−1) compared to the control (77.9 mg ml−1) seedlings at 48h stage (fig. 2A). A similar trend of the result was observed in the ABTS+ assay with enhancement of antioxidant property by UV treatment. Here the sprouts of the 48h stage showed best IC50 values than other stages. The lowest IC50 values i.e. best activity was shown by the sprouts (fig. 2B) with UV 365 nm treatment (16.3 mg ml−1) followed by UV 254 nm (19.36 mg ml−1) and combination of both (29.89 mg ml−1) and in contrast, the highest value was observed in control set (30.8 mg ml−1). FRAP results also exhibited increased in the reducing activity of the sprouts treated with UV light (fig. 2C).

However, the best result was obtained from the seedlings treated with 365 nm which showed consistently enhanced activity till 96h of germination time. Similar to DPPH and ABTS assay the result of metal chelating capacity also showed that the chelation capacity was enhanced best at the 48h stage than other germination stages (fig. 2D).
Fig. 2: A) DPPH and B) ABTS radical scavenging, C) reducing power and D) metal chelating activity of the fenugreek sprouts of 24h, 48h, 72h and 96h stage, untreated and treated with UV radiation of wavelength 254 nm, 365 nm and combined (254 nm+365 nm). Results are represented as mean±SEM, n=3. Values with different letters (a, b, c etc) are significantly (P<0.05) different from each other by Duncan’s multiple range test (DMRT).

In vitro antidiabetic activity

For in vitro antidiabetic activity, the capacity of the extract to inhibit the activity of two key enzymes α-glucosidase and α-amylase was estimated. The inhibition capacity of the sprouts was expressed in terms of IC₅₀ values. The results revealed that the fenugreek sprouts have potential anti-diabetic property and interestingly it was also observed that the UV irradiation resulted in enhancement of this activity in fenugreek sprouts. Both the α-glucosidase and α-amylase inhibition capacity was best enhanced by UV 365 nm at 48h stage though at 24h and 72h stage the activity was considerably increased in comparison to other treatments and the control set (fig. 3).

Fig. 3: In vitro α-glucosidase and α-amylase inhibitory activity of the fenugreek sprouts of 24h, 48h, 72h and 96h stage, untreated and treated with UV radiation of wavelength 254 nm, 365 nm and combined set (254 nm+365 nm). Results are represented as mean±SEM, n=3. Values with different letters (a, b, c etc) are significantly (P<0.05) different from each other by Duncan’s multiple range test (DMRT).

CNT: Control; 254: 254 nm; 365: 365 nm; COMB: Combination of 254 nm and 365 nm

Histochemical analysis

Histochemical analysis was performed to examine the site of localisation of two highly reactive radicals namely superoxide anion and hydrogen peroxide as well as to determine the degree of lipid peroxidation and plasma membrane integrity of fenugreek seedlings. The KI-starch reagent was used to detect the H₂O₂ localisation which imparted blackish stain at the H₂O₂ localised...
The study revealed that the localization of \( H_2O_2 \) molecule was much higher in the seedlings exposed to the UV irradiation than the control one as evidenced by the stain imparted by KI-starch reagent (fig. 4A). For detection of superoxide anion in the tissue, nitro blue tetrazolium chloride solution was used as staining reagent which imparted bluish-black stain. It was observed that UV exposed roots were stained much darker than the control set, thus indicating higher \( O_2^- \) localization (fig. 4B).

The roots were stained with Schiff’s reagent for determining lipid peroxidation as indicated by pink colour. The roots treated with UV radiation exhibited a higher degree of lipid peroxidation thus indicating the oxidative stress created by UV light (fig. 4C). In the case of the plasma membrane integrity similar trend was observed; roots subjected to UV light coloured much darker blue stain with Evan’s blue reagent (fig. 4D) thus exhibiting a higher degree of disintegration of the plasma membrane.

**Fig. 4: Histochemical detection showing the effect of UV exposure on A) \( H_2O_2 \) localization B) superoxide localization C) lipid peroxidation and D) plasma membrane Integrity of the fenugreek seedlings**

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**On gel assay of antioxidative enzymes**

The activity of three different antioxidant enzymes namely superoxide dismutase, catalase and peroxidase were determined by native polyacrylamide gels. On determining SOD activity, five isoforms of SOD was detected in the fenugreek sprouts (fig. 5A). The patterns of the isozymes were different for the control and the UV treated seedlings. The seedlings treated with UV 254 nm did not show a significant change in the activity of the isozymes except for isozyme-II. However, UV 365 nm and combination (254 nm+365 nm) exhibited a remarkable increase in the intensity of three isoforms-I, II and III, also a slight increase in the isoforms-IV and V thus indicating an increase in the SOD activity by UV irradiation. The relative densities of each SOD isoforms are represented in the fig. 5B where the increase in the SOD activity was evident. For catalase, only one isoform were detected and here too the catalase band with greater intensity was found in the sprouts subjected to UV exposure except for UV 254 nm (fig. 6A), thus confirming an increase in catalase activity. The relative density of catalase isozyme is represented in the fig. 6B. For peroxidase also, only single isozyme was found with greater intensity of the band of the sprout treated with UV radiation except for UV 254 nm (fig. 7A). The relative density of peroxidase isozyme is represented in the fig. 7B. Another important enzyme NOX was also assayed by native polyacrylamide gels.

As a result, five isoforms of NOX was detected in the fenugreek seedlings (fig. 8A). The exposure with UV 365 nm resulted in enhanced activity in all the five isoforms, UV 254 nm also enhanced the activity of all isoforms except in isoforms-II the activity was slightly declined and the combined exposure resulted in a higher intensity of all other isoforms except a marginal decrease in isozyme-V. The relative density of NOX isozyme is represented in the fig 8B.
Fig. 6: Effect of UV exposure on the CAT activity of the fenugreek seedlings. A) On gel visualisation of CAT activity of treated and untreated seedlings. B) The relative density of the CAT isoforms as detected by Image lab (Version 5.1, BIORAD) software

Fig. 7: Effect of UV exposure on the POD activity of the fenugreek seedlings. A) On gel visualisation of POD activity of treated and untreated seedlings. B) The relative density of the POD isoforms as detected by Image lab (Version 5.1, BIORAD) software

Fig. 8: Effect of UV exposure on the NOX activity of the fenugreek seedlings. A) On gel visualisation of NOX activity of treated and untreated seedlings, the arrows show the band corresponding to isoforms. B) The relative density of the NOX isoforms as detected by Image lab (Version 5.1, BIORAD) software

Table 1: Pearson correlation coefficients between various phytochemical attributes and antioxidant activity of fenugreek sprouts of different developmental phase

|    | DPPH  | ABTS  | RP    | MC    | TPC   | TFC   | α-glu  | α-amy  |
|----|-------|-------|-------|-------|-------|-------|--------|--------|
| ABTS | 0.653** | -0.298 |       |       |       |       |        |        |
| RP   | -0.132 |       |       |       |       |       |        |        |
| MC   | 0.738** | 0.547* | -0.349|       |       |       |        |        |
| TPC  | -0.745** | -0.643** | -0.099|       |       |       |        |        |
| TFC  | 0.350  | 0.052  | 0.659**|       |       |       |        |        |
| α-glu | 0.562*  | 0.290  | -0.168 | 0.179 |       | -0.429 | 0.372  |        |
| α-amy| 0.607** | 0.337  | -0.039 | 0.251 | -0.588**| 0.517* | 0.924**|

Fig. in bold indicate significant correlation; (*, **) = significant at p<0.05 or 0.01, respectively

**DISCUSSION**

The health benefits of phenolics have been widely reported by various workers. In human beings, phenolic compounds are known to exhibit potential antioxidant property, which is considered to be actively involved in the prevention of oxidative stress-mediated diseases and cardiac disorders. The theory of natural product chemistry suggests that most of the UV-absorbing factors are the active molecules produced by the plants in the form of secondary metabolites [24]. This implies that UV irradiation technology can be implemented in the enhancement of the production of these bioactive molecules for further enrichment of the nutraceutical properties of plant samples.
In the present study, a significant increase in the phenol content of seedlings exposed to UV radiation was found when compared to that of the control set. Similarly, it was observed that phenolic compounds were increased in broccoli (UV-C) and Withania somnifera (UV-B) after exposure to UV radiation [25].

It has been reported that several defensive strategies have been developed by plants to cope with UV radiation. The accumulation of UV-B protective counterparts such as phenols and flavonoids synthesized during phenylpropanoid pathway helps in attenuating the impact of UV radiation [26]. Hence the gradual increase in the flavonoid content at a higher rate in the UV treated seedlings than that of control seedlings can be explained by the higher accumulation of flavonoids as a defence mechanism in response to UV exposure. Our findings are in agreement with that of Gonzalez-Aguilar et al. [27] who reported an increase in flavonoid content in mango, on exposure to UV radiation. Nowadays, flavonoids have gained considerable interest as they are found to be very effective contributors involved in protection against cardiovascular disorder by mitigating the oxidation of LDL as well as prevention of several other degenerative diseases [28]. However, these bioactive compounds (flavonoids) are considered to be absorbed in trace amounts in vivo; thus, there is a general tendency to either increase the consumption or enhance the flavonoid content of plant sources, as this can help in increasing the intake of these essential bioactive compounds and antioxidants.

DPPH, a free radical is used as a simple and economical molecule to determine the antiradical activity of the sample. This method is widely implemented to evaluate the potential of a compound to act as an efficient free radical scavenger or hydrogen donors [11]. The antioxidant activity of the fenugreek sprouts was determined by the ability to scavenge the DPPH radical in the reaction mixture. The scavenging effect of the sample was expressed in terms of IC₅₀ values, thus lower IC₅₀ values indicated higher antioxidant capacity. The results suggested that the sprouts exposed to UV radiation i.e., 254 nm and 365 nm exhibited lower IC₅₀ values than that of control indicating enhancement of antioxidant activity by UV exposure. A similar trend was observed in the case of ABTS assay. A strong positive correlation was observed between ABTS and DPPH assays with a Pearson correlation coefficient of r = 0.653 (table 1).

The reducing capacity of a plant sample acts as a significant indicator of its antioxidant potential. In the reducing power assay, the antioxidants convert the oxidized form of iron (Fe³⁺) from ferric to ferrous (Fe²⁺) [14]. The reducing ability of the sample is expressed in correspondence to ascorbic acid equivalence. The Higher ascorbic acid equivalent value of extract indicated higher reducing capacity. It was observed that the sprouts treated with UV radiation showed enhanced reducing activity with higher ascorbic acid equivalent value when compared to that of control.

Another essential antioxidative mechanism involves the chelation of transition of metal ions such as copper (Cu²⁺) and ferrous ions (Fe²⁺), which leads to prevention of these metal ions to participate in Fenton and Haber–Weiss reactions, ultimately inhibiting the generation of highly reactive hydroxyl radicals which are responsible for oxidative stress-mediated diseases [29]. The antioxidant compounds that are efficiently involved in ferrous ion chelation helps in the mobilization of iron molecules present in tissues by the formation of soluble, stable complexes that are eventually excluded along with excretory products [30]. Hence in recent times, the chelation therapy is widely implemented in the treatment of iron-related complications such as thalassemia which is characterized by overloaded iron in vital body organs. In the present work, the UV exposure resulted in the enhancement of the metal chelating capacity of fenugreek sprouts thus further enriching the nutraceutical property of the sprouts.

A significant correlation was found between the antioxidative properties and the phenolic compounds. The Pearson correlation coefficient obtained between phenol content and the IC₅₀ value of antioxidant assays namely DPPH, ABTS and metal chelating were found to be 0.745, 0.643 and 0.516 respectively (table 1). In agreement with present findings, earlier studies have also reported such significant correlation between phenolics and antioxidant property [29, 31]. Such high correlation between the phenolics and the various antioxidative assays indicates the active participation of these secondary metabolites in the prevention of oxidative stress mediated disorders.

The two very important enzymes of the digestive system, α-amylase and α-glucosidase are considered to be involved in the release of glucose molecules in the blood via starch breakdown. High sugar concentration in blood can cause diabetic complications in a human which directly affects the functioning of internal organs. Hence an inhibition of these enzymes is essential for minimizing the starch breakdown in order to reduce the postprandial hyperglycemic level [32]. The increased level of reactive oxygen species (ROS) has been associated with the degradation of pancreatic beta-cells leading to type 1 diabetes and the onset of type 2 diabetes by insulin resistance [33]. Therefore, the antioxidant compounds which are an efficient scavenger of reactive oxygen species might play important role in controlling oxidative stress mediated diabetic complications. In the present work, the results revealed that the fenugreek sprouts treated with UV radiation at 365 nm exhibited a significant enhancement in the α-amylase as well as α-glucosidase inhibitory activity over control and other treated sprouts. Also, it was observed that the same set of sprouts (exposed to 365 nm) had higher phenolic and flavonoids content. Similar to our findings, various other workers have also reported that phenolic-rich extract has shown the better capability to inhibit α-amylase and α-glucosidase activity [34, 35]. Interestingly, the Pearson correlation coefficient analysis also revealed a highly positive correlation of α-amylase and α-glucosidase inhibitory activity with DPPH based antioxidant activity also with those of the phenolic contents of the sprouts. Polyphenolic compounds have long been recognized as a potent inhibitor of the digestive enzymes because of their capability to bind with these proteins. Several other authors have also shown the carbohydrate hydrolyzing enzyme inhibitory activity of the phenolic compounds in various plant samples [35]. Thus the enhancement in antioxidant and anti-diabetic activity can be attributed to the increased in the polyphenolic content in response to UV exposure.

The major reactive forms of oxygen which mainly exist in the form of Singlet oxygen, superoxide radicals, hydrogen peroxide and hydroperoxide are considered to be the molecules with greater toxic potentials to plant tissues. UV radiation is known for the increase in production of these reactive oxygen species in cell system [36]. As a result, it was observed that the roots of the experimental sets which were treated under UV exposure showed significant accumulation of H₂O₂ and superoxide in the roots and showed darker stain than control thus, indicating that UV radiation leads to increase in production of these reactive molecules. The accumulation of such reactive oxygen molecules has been found to damage the membranes of the cell and organelles which leads to a loss in their functions. The damages and injuries of biological membranes due to peroxidation of polyunsaturated fatty acids are generally correlated with the accretion of malonaldehyde [18] which is also a marker of lipid peroxidation process. As a consequence, the membrane fluidity is altered and ultimately ion leakage occurs from the tissue. In agreement to an aforesaid statement, it was observed that the roots of the seedlings exposed to UV radiation reflected higher malonaldehyde ([dark stain with Schiff’s reagent] and imparted darker stain against Evan’s blue thus indicating loss of plasma membrane integrity. Similar to our findings other workers have also reported a higher level of lipid peroxidation in cucumber leaves due to UV irradiation [37].

Plants have evolved various protective mechanisms for minimising such deleterious effects of these free radicals. Among these defence strategies, both enzymatic and non-enzymatic mechanisms are involved. The enzymatic defence comprises of the efficient antioxidant enzymes such as catalase, peroxidase, superoxide dismutase, polyphenol oxidase etc. [38]. In the present study, the native on gel analysis was performed for two important antioxidant enzymes namely, catalase and superoxide dismutase. Studying ROS metabolism related to UV exposure includes investigation of enzymatic components of the antioxidant system. It has also been reported that UV irradiation leads to enhancement in activity of the antioxidant system, as a defence mechanism [39].
The results obtained in the present study indicates a considerable increase in the catalase activity of the sprouts subjected to UV irradiation (UV 365 nm and Comb) which was clearly visible on the gel with an increase in the intensity of the catalase isozyme of UV treated sprouts. A similar trend was observed in the case of peroxidase isozyme pattern with seedlings exposed to UV 365 nm and Comb, showing best result. The peroxidase enzymes are reported to utilise phenolic compounds as their co-substrate and this fact was further confirmed in the present work by the presence of high phenolic content in the sprouts treated with UV 365 nm. Elevation in the catalase and peroxidase activity has also been observed by other workers in different plant samples under UV irradiation [40]. It can be suggested that the excess H$_2$O$_2$ production might have resulted in induced catalase as well as peroxidase activity in response to UV exposure.

It was observed that the intensity of the SOD isozymes was found to increase in the lane loaded with protein extracted from seedlings subjected to UV irradiation, which indicates an increase in the antioxidative activity of SOD enzyme. Also, the increase in the SOD activity might be attributed to the production of superoxide radical due to UV irradiation as earlier suggested by [41] in leaves of barley cultivar. Such concomitant elevation with SOD activity and removal of H$_2$O$_2$ resulted from dismutation of O$_2$· offers complementary role of catalase with SOD activity.

Plasma membrane-bound NOX is considered to be involved in the regulation of the antioxidative defence mechanism during stress condition in plants [42]. Therefore, in the present study, the NOX isozyme analysis was performed on a native gel and the obtained result demonstrated an increase in the intensity of the NOX isozyme bands, thus indicating overexpression of NOX during UV irradiation. NOX has been known to be responsible for the production of superoxide anion radical. Thus, it can be stated that the increase in NOX activity leads to increase in O$_2$ production which further activates SOD activity which in turn dismutates O$_2$ to H$_2$O$_2$ which influxes in the cellular space further activating catalase and peroxidase to overcome the ROS production for removal of H$_2$O$_2$ molecules. Enhancement in the activity of these antioxidant enzymes might be an acclimatization mechanism by the fenugreek seedlings to overcome oxidative stress imposed by UV irradiation.

CONCLUSION

In conclusion, when subjected to the UV exposure the fenugreek sprouts showed enhancement in the free radical scavenging capacity as well as anti-diabetic activity. It can be suggested that the induction of NOX enzyme by UV exposure which is responsible for the production of O$_2$ has resulted in the activation of SOD, which further dismutates O$_2$ to H$_2$O$_2$. The increase in the production of O$_2$ and H$_2$O$_2$ has been evident in the histochemical detection, thus increase in H$_2$O$_2$ concentration might resulted in overexpression of catalase and peroxidase activity. Since peroxidase is known to utilise phenolic compounds as a substrate, the observed enhancement of peroxidase activity might also be due to increase in phenolic contents in the sprouts during UV exposure. Further, it can be explained that the increase in the phenolic compounds was found to be responsible for the enhancement in antioxidant and antidiabetic properties of sprouts. These findings suggest that UV irradiation technology can be used as a simple and cost effective method for the production of sprouts with enriched nutraceutical properties.

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CONFLICT OF INTERESTS

Authors claim no conflict of interest.

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