Cytotoxic and antioxidant properties of active principals isolated from water hyacinth against four cancer cells lines

Ahmed M Aboul-Enein¹, Sanaa MM Shanab², Emad A Shalaby¹*, Malak M Zahran³, David A Lightfoot⁴ and Hany A El-Shemy¹

Abstract

Background: *Eichhornia crassipes* (Mart) solms is an invasive macrophyte causing serious problems to the network of irrigation and drainage canals in the Nile Delta region. The present study aim to evaluate the potential anticancer and antioxidant activities of *Eichhornia crassipes* crude extract and its pure compounds.

Methods: The macrophyte was collected from El-Zomor canal, River Nile (Egypt), cleaned, air dried, grinded then extracted with methanol (crude extract). The extract was fractionated using pre-coated silica gel plates (TLC F₂₅₄) with hexane/ethyl acetate (8.5: 1.5 v/v) as mobile phase. Nine fractions were separated (A-I) then scratched, eluted with the same mobile phase, filtered and the separated fractions were determined and identified using spectroscopic methods (Mass spectrum (MS), Infra red (IR) and Proton H-Nuclear magnetic resonance (H-NMR). Both the crude extract and its nine identified compounds were tested for their antioxidant (using 2, 2′- diphenyl-1-picrylhydrazyl (DPPH), 2, 2′- azino-bis {ethylbenzthiazoline-6-sulfonic acid (ABTS.)} methods) and anticancer activity (using MCF-7, HeLa, Hep.G2 and EACC cell lines).

Results: The antioxidant and anticancer activities of the crude extract exhibited the highest effect while the compounds showed variable effects which depend on the type of compound and cancer cell line. The antioxidant activity of the crude extract exhibited the highest followed in descending order by compounds D, E, G and H respectively. Concerning the anticancer potency, the crude extract showed also the highest effect while the identified compounds (A, B, C, D, E, F, G, H and I) showed variable anticancer activities against the four different cell lines. In addition, Compound I exhibited the most potent anticancer activity against HepG2 cell line while compound D exhibited high anticancer activity against HeLa cells and EACC. The results revealed the presence of different compounds (Alkaloids and terpenoids) with variable antioxidant and anticancer activities which elicited an auto-augmentation in the crude extract leading to its greatest activities. The action of the identified anticancer compounds on DNA fragmentation was studied.

Conclusion: The study illustrated the potential of *Eichhornia* as a valuable resource for natural compounds of desirable medicinal properties (e.g. antioxidants and anticancer).

Keywords: *Eichhornia crassipes*, Cytotoxicity, Anticancer, Antioxidant, Active ingredients
Background
The role of plants in maintaining human health is well documented [1]. Plants are used medicinally in different countries and are a source of many potent plus powerful drugs [2]. The active principles of many drugs found in plants were identified from secondary metabolites [3]. Amongst these active phyto-molecules (at least 12,000) about only 10% were chemically identified [4]. The medicinally useful bioactive constituents were belonged to alkaloids, flavonoids, phenolics, essential oils and polyphenols [5]. Alkaloids were play an important metabolic role and control development in living systems [6]. These chemical substances were known to carry out important medicinal roles in human body. They were also involved in protective function in animals and used as medicine especially the steroidal alkaloids [7].

Eichhornia crassipes (Mart) Solms, commonly known as water hyacinth, is warm aquatic plant belonging to the family Pontideraceae. Water hyacinth was listed as one of the most productive plants on earth and considered the world’s worst aquatic plant [3]. It originated in the state of Amazones, Brazil, spread to other regions of South America and was carried by humans to tropical and subtropical regions. It invaded Africa and causes extremely serious ecological, economical and social problems in the region between 40 degree north and 45 degree south [8].

The dense mats of water hyacinth cover lakes and rivers, blocking water ways and interfering with the water transport of agriculture products, tourism activities and irrigation of agriculture fields. Its vast growth consumes huge quantity of nutrients favoring and its growth over other aquatic species [9,10]. Decomposition of dead individuals released nutrients to water leading to deterioration of water quality, clean drinking water which threatened human health [9,10]. In summer season where temperature exceeded 35°C and high solar radiation with long light duration, Eichhornia grows faster with rapid spread in the Egyptian water bodies and serious problems occurred [9,10].

Epidemiological studies reveal that, numbers of diseases including cancer, diabetes, Alzheimer, coronary heart disease and aging have been found to be associated with oxidative stress and reactive oxygen species (ROS) [11] which damage biomolecules and lead to numerous disease conditions [12]. Dietary antioxidants may reduce the risks of these diseases and improve general human health [13,14]. Besides their effective role in prevention of diseases, antioxidants were also used as food additives to improve quality of food products by avoiding offensive flavors [15]. Naturally produced antioxidants including flavonoids, tannins, coumarins, xanthones, phenolics, terpenoids were recorded in various plant products (fruits, leaves, seeds and oils) and organisms [9,10].

Recently, considerable attention has been given to harvest E. crassipes for practical uses as economical sources in many parts of the world. Though, very few pharmacological studies on this plant have been reported in literature and are only used as therapeutic properties in folkloric application [16]. Our previous studies showed promising antimicrobial and anti-algal activities of water hyacinth extract and some of its identified compounds [17] encouraged us to go further for other biological effects of this plant. Therefore, the present study aimed to isolate and evaluate the potentially active substances or fractions with potential pharmacological activity of water hyacinth. This investigation was carried out on crude extract of this weed as well as its fractions for potential anticancer and antioxidant activities. The fractions were chemically identified and their action on cancer cells was elucidated.

Methods
Chemicals and drugs
Pure hexane, chloroform, ethanol, ether, acetone, methanol and acetic acid were purchased from E. Merck Co. (Darmstadt, Germany). Sulfarhodamine, 2, 2 diphenyl-1-picrylhydrazyl (DPPH), 2, 2′- azino-bis (ethylbenzthiazoline-6-sulfonic acid (ABTS +) were purchased from Sigma-Aldrich (St. Louis, MO, USA). Trichloroacetic acid and other materials were of the highest available commercial grade.

Collection and extraction of water hyacinth
E. crassipes (Mart) Solms was collected during spring season (2010) from the River Nile at El-Zomor canal (Giza), Egypt and was identified by Dr. Sanaa M. M. Shanab, Professor of Phycology, Botany and Microbiology Department, Faculty of Science, Cairo University. This species has been deposited in a scientifically available herbarium (Botany and Microbiology Department, Faculty of Science, Cairo University).

It was cleaned from any debris, washed several times with tap and distilled water then with sterilized water. The samples were air dried (and/or lyophilized), grinded and then stored at -20°C until use. A known weight (500 g) of the macrophyte was extracted three times (Using 5 liter solvents with 24 h intervals) with methanol at 25°C in the dark. The solvent of the combined extracts was evaporated, and solids concentrated using rotary evaporator at 40°C and weighted.

Fractionation of the crude extract
Using pre-coated TLC F254 plates, a solution of the crude methanolic extract (Ten grams) containing a mixture of compounds is applied to the layer of adsorbent, near one edge, as a thin line. The TLC plate is dropped vertically in a closed container (developing chamber), with the edge to which the spot was applied down. The solvents mixture, which is in the bottom of the container, travels up the layer of adsorbent by capillary action, passes over the
spot and, as it continues up, crude extract was fractionated using different solvent systems as mobile phase and the suitable one was selected (hexane/ethyl acetate). In this concern, different combinations of hexane/ethyl acetate solvents (9:1, 8.5:1.5, 8:2, 7:3 and 5:5) as the mobile phase were prepared and tested. The compounds were moved using hexane/ethyl acetate (8.5:1.5, v/v), as the best solvent The separated nine fractions (A-I) by the mixture, were scratched and eluted with the same mobile phase then filtered, evaporated (using nitrogen gas), weighed and used for chemical identification and in bioassays experiments [17].

Instrumental analysis
Potent fractions were chosen for further identifications using the chromatographic and spectroscopic methods as follows:

a- Mass spectroscopic (MS) analyses of potent fractions
The biologically potent fractions of *E. crassipes* were analyzed by Mass spectra (MS). The mass spectrum was scanned over the 40 to 500 m/z range with an ionizing voltage of 70 eV and identification was based on standard mass library of National Institute of Standards and Technology (NIST Version 2.0) to detect the possible fraction structure.

b- Fourier transformed infra red (FTIR) spectra
A Perkin Elmer (Waltham, Massachusetts, USA) was used to obtain Fourier transformed infrared (FTIR) spectra (System, 2000) of the fractions.

c- Proton nuclear magnetic resonance Spectra (1H NMR)
The identification of fractionated compounds was confirmed by carrying out 1H-NMR analysis using NMR Joel GIM, EX 270 (400 Hz).

Assessment of antioxidant activity
Two methods were used for antioxidant evaluation of crude extract as well as its active ingredients as follows:

a- DPPH method
The 2, 2 diphenyl-1-picrylhydrazyl (DPPH) tests were carried out for evaluation of antioxidant activity of samples as described by Burits and Bucar [18]. One mL of *Eichhornia* crude extract and fractions at different concentration (From 5 to 500 μg/mL) was mixed with 1 mL DPPH reagent [0.002% (w/v)/ methanol solution]. After an incubation period (30 min), the absorbance was measured at 517 nm. Extract concentration providing 50% inhibition (IC<sub>50</sub>) was calculated from the graph plotting inhibition percentages against extract concentrations compared with Butylated hydroxy anisole (BHA) and Butylated hydroxy toluene (BHT) as synthetic antioxidant and Ascorbic acid (Vit. C) as natural antioxidant standard.

\[
\text{% Antioxidant activity } (E) = \left( \frac{\text{Ac} - \text{At}}{\text{Ac}} \right) \times 100
\]

Where: Ac was the absorbance of methanolic DPPH solution (control) and At was the absorbance of the extract sample.

b- ABTS<sup>•</sup> radical cation scavenging assay
This assay was based on the ability of different fractions to scavenge 2, 2'- azino-bis (ethylbenzthiazoline-6-sulfonic acid (ABTS<sup>•</sup> ) radical cation in comparison to a standard ( BHA, BHT and Ascorbic acid). The radical cation was prepared by mixing a 7 mM ABTS stock solution with 2.45 mM potassium persulfate (1/1, v/v) and leaving the mixture for 4-16 h until the reaction was completed and the absorbance was stable. The ABTS<sup>•</sup> solution was diluted with ethanol to an absorbance of 0.700 ± 0.05 at 734 nm for measurements. The photometric Assay was conducted on 0.9 mL of ABTS<sup>•</sup> solution and 0.1 mL of tested samples (in MeOH solution) and mixed for 45 s; measurements were taken immediately at 734 nm after 1 min. The antioxidant activity (E) of the tested samples was calculated by determining the decrease in absorbance at different concentrations by using the following equation:

\[
\text{% Antioxidant activity } (E) = \left( \frac{\text{Ac} - \text{At}}{\text{Ac}} \right) \times 100
\]

Where At and Ac are the respective absorbance of tested sample and ABTS<sup>•</sup>, [19]. Extracts concentration providing 50% inhibition (IC<sub>50</sub>) was calculated from the graph plotting inhibition percentages against extract concentration.

Assessment of anticancer activity
a- Cell culture
Human hepatocellular cancer cell line (HepG-2), breast cancer cell line (MCF-7) and cervix cancer cell line (HeLa) were obtained from the Vacsera (Giza, Egypt). Cells were maintained in RPMI-1640 supplemented with 100 μg/mL streptomycin, 100 units/mL penicillin and 10% heat-inactivated fetal bovine serum in a humidified, 5% (v/v) CO<sub>2</sub> atmosphere at 37°C [20]. Ehrlich Ascites Carcinoma Cells, EACC were obtained from National Cancer Institute, NCI, (Cairo, Egypt) and cell line was maintained in female albino mice by i.p transplantation of 0.2 mL ascites which contains about 10<sup>6</sup> of tumor cells in a liquid form and cells were taken for study after 15-20 days of tumor transplantation.
b. Cytotoxicity assays

1-Against solid tumor cell line

The cytotoxicity of crude extract and the fractionated compounds were tested against HeLa, HepG-2 and MCF-7 cells by SRB assay as previously described [21] or by neutral red assay [22]. Exponentially growing cells were collected using 0.25% Trypsin-EDTA and plated in 96-well plates at 1000-2000 cells/well. Cells were exposed to each test compound for 72 h and subsequently fixed with TCA (10%) for 1 h at 4°C. After several washings, cells were exposed to 0.4% SRB (sulforhodamine B (SRB), 2-(3-diethylamino-6-diethylazaniumylidene-xanthen-9-yl)-5-sulfo-benzenesulfonate) solution for 10 min in dark place and subsequently washed with 1% glacial acetic acid. After drying overnight, Tris-HCl was used to dissolve the SRB-stained cells and color intensity was measured at 540 nm. Doxorubicin (DOX) purchased from Sigma Chemical Co. (St. Louis, MO, USA) was used as anticancer standard.

Data analysis

The dose response curve of compounds was analyzed using E_{max} model (Eq. 1).

\[
\% \text{ Cell viability} = (100 - R) \times \left(1 - \frac{|D|^m}{K_d m + |D|^m}\right) + R
\]

Where R is the residual unaffected fraction (the resistance fraction), |D| is the drug concentration used, K_d is the drug concentration that produces a 50% reduction of the maximum inhibition rate and m is a Hill-type coefficient. IC_{50} was defined as the drug concentration required to reduce fluorescence to 50% of that of the control (i.e., K_d = IC_{50} when R = 0 and E_{max} = 100-R) [22].

2-Against liquid tumor cell line

The cytotoxicity of crude extract and its pure fractions was also tested against EACC by viability trypan blue exclusion test according to previous report [20]. In details; A line of Ehrlich Ascites Carcinoma resistant to endoxan has been used. The parent line was first supplied through the courtisy of Dr. G. Klein, Amestrdam, Holland. The tumor line is maintained in the National Cancer Institute, Egypt in Female Swiss Albino Mice by weekly transplantation of 2.5×10^6 cells were centrifuged at 1000 x g for 5 min at 4°C. The pellets were washed with saline (0.9% NaCl) then the needed number of cells was prepared by suspending the cells in the appropriate volume of saline. The viability percentage (V%) of tumor cells was measured after incubation with the tested extract and fractions as well as DMSO as control. Two mL of cells (4×10^6 cells) were transferred into a set of tubes, then different samples at different concentrations (5 to 500 μg/mL) were added into the propitiate tubes as well as DMSO. The tubes were incubated at 37°C for 2 h. Then, in a test tube containing 80 μl saline and 10 μl trypan blue, 10 μl of cell suspension were added and mixed then the number of living cells was calculated under microscope using a hemocytometer slide.

DNA extraction and electrophoresis

DNA of cancer cell lines before and after incubation with crude plant extracts and each of the active identified ingredients was extracted using AXYGEN Biosciences (USA) extracted kit. The extracted DNA was precipitated by ethanol and separated by centrifugation at 12000 rpm (at 4°C) for 15 min and the pellets were purified. The purified DNA was electrophorised using agarose gel electrophoresis at 150 V for about 90 min to assess the degree of apoptotic DNA fragmentation according to Gao et al., [23]. The DNA and its fragments on gel were detected by treatment with ethidium bromide and the florescence produced by UV light was photographed and analyzed using gel documentation system (G box, syngene, UK).

Hypochromicity evaluation

This experiment was done to evaluate the possibility of active ingredients (A-I) to interchelate with DNA (isolated from EACC cell line), whose absorption decreases upon binding at 260 nm (term called hypochromicity). The comparison was done against free DNA and the tested compounds.

Statistical analysis

Data were subjected to an analysis of variance, and the means were compared using the “Least Significant Difference (LSD)” test at 0.01 levels, as recommended by Snedecor and Cochran [24]. Data are presented as mean ± SD.

Results and discussion

Active ingredients

Nine fractions were separated and identified from water hyacinth crude extract (Figures 1 and 2) and showed biological activities as antioxidant and anticancer. The obtained data of the spectroscopic analysis revealed the nine compounds as; the alkaloid derivatives (A, D and F), terpenoids derivatives (B, C, E, G, H and I). The more or less comparable activities of the fractionated compounds from methanolic crude extract of E. crassipes may be due to the fact that all compounds have the common fragment ions: 57, 71, 85, 149, 167 Da and these results were confirmed by H^1-NMR. The NMR data indicated that all compounds had the following type of protons; A multiplex signal at δ 6.89-7.47 ppm was characteristic of aromatic
protons (Compounds A-I) according to Edwards [25];

While, the singlet signal at $\delta$ 5.320 ppm was characteristic of the two protons of olifinic (CH = CH) in compound A; the singlet signal at $\delta$ 3.342 ppm was characteristic of four protons of two O-CH$_2$ group (Compounds B, C, D, E, F and I); the singlet signal at $\delta$ 1.26, 1.6 and 2.5 ppm was characteristic of the protons of methylene (CH$_2$) group and the singlet signal at $\delta$ 1.9 ppm was characteristic of protons of methyl (CH$_3$) group.

Antioxidant activity

The DPPH scavenging assay was performed to test the antioxidant activity (as IC$_{50}$) of both the crude methanolic extract and the nine TLC separated fractions (compounds) of _Eichhornia crassipes_ (Table 1). The crude methanolic extract showed the highest antioxidant activity with IC$_{50}$ of 74.80 ± 4.5 μg/mL. Three of nine compounds recorded more or less lower and comparable activities i.e. 92.40 ± 6.5, 95.4 ± 3.1 and 96.5 ± 2.7 μg/mL of compounds D, E and H. Additionally, compounds B, C, F, G and I were illustrated very close antioxidant effects with IC$_{50}$ ranged between 97.0 ± 5.4 and 97.4 ± 2.7 μg/mL. However, compound A showed very low effect (99.2 ± 5.8 μg/mL) as antioxidant compared with synthetic (BHA and BHT) and natural (Vit. C) Standards.

Activity was decreased on separation of the fractions in pure forms (compounds) and each more or less was lower than that of the crude extract. The potent antioxidant activity manifested by the crude extract in comparison to those of the separated compounds may be due to the synergistic action of the collective biologically active compounds of one or more of the nine fractions in the crude extract. In addition, the crude extract may have secondary metabolites in very low concentration which enhance the active principals and increase the antioxidant efficiency. This suggestion was previously confirmed by Chu et al [26] - Who mentioned that crude extract of _Spirulina_ gave higher antioxidant activity than pure compounds (phycocyanin). They reported that the extract might contain other constituents (e.g. phenolic compounds) which gave a higher combined antioxidant activity than phycocyanin alone. The synergistic action of a wide spectrum of antioxidants may be more effective than the activity a single antioxidants.

It is difficult to give a definite explanation for mechanism of antioxidant of _Eichhornia_ purified compound dependent on one assay. Thus, radical scavenging properties of the pure compounds (A-I) were evaluated against ABTS radical assay, and the results were compared with those of some standard antioxidant (BHT, BHA and Ascorbic acid) (Table 1). It was observed that those compounds showed a markedly higher ability to scavenge ABTS radicals than DPPH. The obtained data showed the trend concerning the crude extract as well as tested compounds. Crude extract scavenged the ABTS radical with IC$_{50}$ values of 50.8 μg/mL. These values were better than the IC$_{50}$ of DPPH radical (74.8 μg/mL) and other fractions but less than standard compounds (BHA, BHT and Ascorbic acid; 9.4, 5.0 and 12.0 μg/mL respectively). This means that, the compounds were strongly scavenged the ABTS radical than that of DPPH radical. These results are in agreement with those of Awika et al. [27] who found that, ABTS is a better choice than DPPH and more sensitive than DPPH. The ABTS method has the extra flexibility in that it can be used at different pH levels (unlike DPPH, which is sensitive to acidic pH) and thus is useful when studying the effect of pH on antioxidant activity of various compounds. It is also useful for measuring antioxidant activity of samples extracted in acidic solvents. Additionally, ABTS is soluble in aqueous and organic solvents and is thus useful in assessing antioxidant activity of samples extracted in acidic solvents. The ABTS method has the extra flexibility in that it can be used at different pH levels (unlike DPPH, which is sensitive to acidic pH) and thus is useful when studying the effect of pH on antioxidant activity of various compounds. It is also useful for measuring antioxidant activity of samples extracted in acidic solvents. Additionally, ABTS is soluble in aqueous and organic solvents and is thus useful in assessing antioxidant activity of samples extracted in acidic solvents.
The antioxidant activity of active ingredients separated from *E. crassipes* may be correlated with the presence of hydroxyl group and unsaturated bonds in the chemical structure of its compounds which show high ability for scavenging free radicals and prevent the oxidation processes. These observations were in agreement with the previously published results [28].

ROS are products of a normal cellular metabolism and play vital roles in the stimulation of signaling pathways in plant and animal cells in response to changes in intra- and extracellular environmental conditions [29]. Proteins, nucleic acids and lipids were also significant targets for oxidative attack, and modification of these molecules can increase the risk of mutagenesis [30]. Therefore, antioxidants are good scavengers for ROS and free radicals, on other word it defends and protect cells from their bad action. The antioxidant property of *Eichhornia crassipes* is very much depend on concentration used (1-1000 μg/mL).

**Antitumor activity**

SRB assay was used to assess the cytotoxicity of the crude extract and its derived fractions against three different solid tumor cell lines and one ascites tumor cell line. Different cell lines were used according to their origin and morphology as well as sensitivity and receptor sites behavior. The cytotoxicity parameter, IC<sub>50</sub>, was calculated using E<sub>max</sub> model as described in the Methods section. Compound is considered strongly potent when IC<sub>50</sub> is lower than 1 μg/mL; moderate if IC<sub>50</sub> ranges from 1 μg/mL to 10 μg/mL; and weak for IC<sub>50</sub> higher than 10 μg/mL. The obtained results of the crude extract showed acceptable potency against HeLa and MCF-7 cell lines with IC<sub>50</sub> of 1.6 ± 0.5 and 1.2 ± 0.2 μg/mL, respectively. However, HepG2 and EACC cell lines showed relatively higher resistance against the crude extract with IC<sub>50</sub> of 7.6 ± 1.5 and 6.04 ± 0.5 μg/mL respectively, when compared with Doxorubicin (DOX) as standard anticancer drug (with IC<sub>50</sub> = 0.28, 0.42 and 0.42 μg/mL against Hela, HepG2 and MCF-7 respectively) as shown in Table 2. This means that the cytotoxicity pattern of the crude extract on both MCF-7 and HeLa cell lines is similar while differs on HepG2 and EACC (Figure 3). These results indicate that the effect of crude extract on MCF-7 and Hela cells is concentration dependant through the concentrations tested (1-1000 μg/mL). In HepG2 cells however,
most potent cytotoxic profile among all other fractions with IC\textsubscript{50} of 0.8 ± 0.4 µg/mL (in HepG2 liver cancer cell line). Compound D showed also high potent cytotoxicity against HeLa cervix cancer cell line (IC\textsubscript{50} = 4.3 ± 2.3 µg/mL) and the other compounds showed moderate cytotoxic effect with IC\textsubscript{50}'s ranging from 7.7 to 14.1 µg/mL. With respect to MCF-7 breast cancer cell line, compounds H, B and A showed the best cytotoxic profile with IC\textsubscript{50} of 11.1 ± 6.1 µg/mL, 13.4 ± 1.9 µg/mL and 13.6 ± 5.3 µg/mL, respectively. In addition, the other compounds G and C showed mild cytotoxic effects with IC\textsubscript{50}'s ranging from 17.5 to 69.1 µg/mL. The other compounds showed much humble cytotoxicity profile against HepG2 cell line with IC\textsubscript{50}'s ranging from 14.9 to 74.2 µg/mL. Concerning EACC cancer cell line, compounds D, C, E, A showed high potency with IC\textsubscript{50} of 6.42 ± 0.8, 7.29 ± 1.6, 8.19 ± 1.2 and 8.61 ± 2.1 µg/mL, respectively (Table 2). The other compounds showed moderate effect with IC\textsubscript{50} of 12.32 ± 2.7, 12.67 ± 4.2 and 22.79 ± 6.9 µg/mL for compounds H, G and I, respectively. The effect of plant extract and its compounds on EACC is very similar to their effects on HeLa cells (Table 2).

It's worth mentioning, that the higher potency of the crude extract against cancer cell lines in special HeLa and MCF-7 relative to all fractions (compounds) from the same extract, might be attributed to auto-synergetic effect of these fractions within the same extract. In addition, the crude extract may have additional secondary metabolites in very low concentration which may enhance the active principals and increase the anticancer efficiency in addition to its stability against enzyme action of tumor cells as mentioned in antioxidant activity [26].

The similar cytotoxic profile of the crude extract against both HeLa (cervix cancer) and MCF-7 (breast cancer) cell lines followed by EACC might be attributed to their gynecological origin and hormone dependency [33]. Several gynecological solid tumors such as breast and cervix cancers are greatly influenced by hormone level and signaling within tumor micro milieu [34,35]. Further studies are recommended to identify potential interference between the isolated potent fractions and estrogenic signaling pathway within gynecological tumors.

In liver cancer cell line, HepG2, the only compound with promising potency was compound I, identified as iso-octyl phthalate (IC\textsubscript{50} = 0.8 ± 0.4 µg/mL). Similar trend was noticed for HeLa and EACC cell lines hence, compound D (methyl dioctyl phthalate) showed high potency as anticancer. This information recommends further SAR and QSAR studies to identify the exact target receptor for these active compounds. Herein, it is interesting to notice that most of the detected phthalate esters have a high potency against cancer cell lines. The longest side chain-ester of phthalate (compound D) has the strongest anticancer activity against HeLa and EACC cell lines. In this

| Materials     | DPPH method | ABTS method | Relative percentage to Vit. C |
|---------------|-------------|-------------|-----------------------------|
| Crude extract | 74.8 ± 4.5\textsuperscript{a} | 50.8 ± 2.4\textsuperscript{a} | 24.64 |
| Compound A    | 99.2 ± 5.8\textsuperscript{a} | 77.0 ± 3.2\textsuperscript{a} | 51.5 |
| Compound B    | 97.1 ± 3.9\textsuperscript{a} | 78.5 ± 2.4\textsuperscript{a} | 15.29 |
| Compound C    | 97.4 ± 2.7\textsuperscript{a} | 77.6 ± 1.3\textsuperscript{a} | 15.47 |
| Compound D    | 92.4 ± 6.5\textsuperscript{a} | 73.2 ± 4.2\textsuperscript{a} | 16.39 |
| Compound E    | 95.4 ± 3.1\textsuperscript{a} | 75.8 ± 2.6\textsuperscript{a} | 15.84 |
| Compound F    | 97.2 ± 4.6\textsuperscript{a} | 77.9 ± 1.9\textsuperscript{a} | 15.40 |
| Compound G    | 97.2 ± 1.6\textsuperscript{a} | 75.8 ± 1.6\textsuperscript{a} | 15.84 |
| Compound H    | 96.5 ± 2.7\textsuperscript{a} | 69.8 ± 3.0\textsuperscript{a} | 17.72 |
| Compound I    | 97.0 ± 5.4\textsuperscript{a} | 76.5 ± 3.1\textsuperscript{a} | 15.62 |
*Standard antioxidants.

Each value is represented as mean of triple treatments, means within each row with different letters differ significantly at P < 0.01 according to Duncan's multiple range test.
concern, Yamamoto et al. [36] reported that phthalate derivatives (compound B, C, D, E and I) had high ability for inhibition of skin tumor promotion induced by 12-O-tetradecanoyl-phorbol-13-acetate. In addition, the present data showed that compound D presented the lowest percentage in the crude extract (only 6%, Figure 4) but has the highest antioxidant and anticancer activity. In contrast, compound I represented the highest percentage fraction (more than 17%) in the crude extract (Figure 4) and showed the strongest anticancer activity against Hep-G2 and EACC respectively (Table 2).

The anticancer activities of compounds A, G and H were not recorded in any other investigation concerned with neither Eichhornia crassipes nor other plants (its first record in this study). The later two compounds (G and H) seem to have potent activity for the four cancer cell lines tested in spite of the fact that they came in the second order of anticancer activity. These compounds constitute only about 22.5% (relative) of the crude extract (G 8% and H 14.5%, Figure 4) which may be more potent by increasing their contents in the media. Therefore, experiment was done to confirm this probability hence, incubation of cancer cells with higher concentrations of each compounds showed higher anticancer activity (EACC was used, data not shown). Also using mixture of the two compounds gave the most potency. In spite of occurring compound G in crude extract in lower percentage but it showed high anticancer activity for the four cell lines used. In this respect, the two compounds are very similar in their chemical structure (formulas), hence they contain biphenyl linked through methylene group and the later carries methyl group. This structure may enhance liberation of methyl ion as carbonium ion which affects DNA of cancer cells [37] or cell proteins and/or enzymes. This liberated methyl ion group can bind with N7- guanine base (or other bases) in DNA and induce mutation which in turn stop cell life cycle or kill these cancer cells [38]. This phenomenon resulted in significant intensities of internucleosomal DNA fragmentations, evidenced by the formation of a DNA laddering on agarose gel (Figure 5), hallmark of cells undergoing apoptosis. This means that nuclear DNA was degraded to different fragments of about 200 bp or its folds which appeared as ladder. This appeared clearly by crude extract treatments and compound I followed by

| Compounds | O.D 260 nm | O.D 260 nm | Expected O.D | Observed O.D After 2 h | Observed O.D After 24 h |
|-----------|------------|------------|--------------|------------------------|------------------------|
| A         | 0.0325     | 0.732      | 0.289        | 0.391                  |
| B         | 0.139      | 0.838      | 1.032        | 1.196                  |
| C         | 0.065      | 0.764      | 0.893        | 1.168                  |
| D         | 0.0995     | 0.7985     | 0.906        | 1.042                  |
| E         | 0.0725     | 0.6995     | 0.863        | 0.99                   |
| F         | 0.3085     | 1.0075     | 1.036        | 1.050                  |
| G         | 0.1245     | 0.8235     | 0.956        | 1.07                   |
| H         | 0.1905     | 0.8895     | 1.080        | 1.04                   |
| I         | 0.0850     | 0.7840     | 0.841        | 0.928                  |
compounds H and A and then F and E (lanes 10, 7, 5, 3 and 8 respectively, Figure 5).

The presence of short side chain in compound H (but not compound G) may enhance methyl ion liberation due to their electro negativity potential. The potency of H compound was expected, which showed higher anticancer activity than compound G as noticed in Table 2. Data of the present study also showed that, HeLa cells and EACC have higher sensitivity to natural anticancer (extracts or compounds) than HepG2 and MCF-7 cell lines which showed resistant behavior for the treatments. In this concern, we found that the incubation of these cells with mixture of the high potent compounds (I, F, G and D) gave high anticancer activity and the anticancer effect was concentration dependent.

We did an additional experiment to test the intercalation possibility of each compound with DNA of the cancer cells (detection of hypochromicity through absorption at 260 nm). Results in Table 3 and Figure 6 showed that compound A was the only compound which can intercalate with DNA while the other eight compounds (B-I) were not. The absorbance of DNA + compound mixture was increased (loss of hypochromecity) after 2 and 24 h of mixture which illustrated DNA damage. Therefore, the damage in cancer cells by specific natural compounds did not depend only on their intercalation with DNA but might also to strands cleavage (DNA denaturation) or the base damages [by alkylation and/or free radical attack] or its liberation.

In this concern, we conclude that natural anticancer substances can affect cancer cells through their DNA damage by different probability mechanisms.

**Antioxidant-anticancer relationship**
The present work showed a relationship between the antioxidant activity of natural compounds and their anticancer potency.
In this concern, extensive research during the past 2 decades has revealed the mechanism by which continued oxidative stress can lead to chronic inflammation, which in turn could mediate most chronic diseases including cancer, diabetes, and cardiovascular, neurological, and pulmonary diseases. Oxidative stress can activate a variety of transcription factors including NF-κB, AP-1, p53, HIF-1α, PPAR-γ, β-catenin/Wnt, and Nrf2. Activation of these transcription factors can lead to the expression of over 500 different genes, including those of growth factors, inflammatory cytokines, chemokines, cell cycle regulatory molecules, and anti-inflammatory molecules [38].

On the other side, the role of antioxidant compounds on growth inhibition of cancer cells could be explained as one or more of following suggested mechanisms:

1. The resonance phenomena of double bonds and lone pair atoms (N, S, O) in the chemical structure of the active compound (Figure 2). This structure may lead to radical formation in more than one site; for example: benzene ring (A), this ring is near to highly negativity group (nitro group); this condition helps the benzene ring to convert it to radical formation of new covalent bond with ABTS radical (as antioxidant activity). This radical may react with nitrogen bases of DNA and lead to mutation in nucleic acid of cancer cells and affect cell division (acts as anticancer activity) [39].

2. Normal tissues typically have median oxygen concentrations in the range 40–60 mmHg, half of all solid tumors have median values less than 10 mmHg with fewer than 10% in the normal range. So, the antioxidant compounds may be affecting on cancer cells by providing high concentration of oxygen by acting with its radicals form and in turn disturbs tumor hypoxia needed for these cells [35].

3. Recent studies have demonstrated that antioxidant compounds (including selenium) are also essential elements of redox system, which has been shown to have multiple functions, including regulating cell growth and apoptosis [2,3]. Selenium may exert its effects in cancer cells by altering intracellular redox state, which subsequently results in cell cycle block. In addition, these compounds have been demonstrated to possess antitumorigenic activities in animal models by inhibiting tumor initiation and promotion through disturbing redox potential system of tumor cells. This takes place via modifying cellular antioxidants and antioxidant enzymes, thus regulating the therapeutic effectiveness of natural drugs in cancer therapy [40].

4. As recommendation, the addition of antioxidant compounds with anticancer chemical drug enhances the action of chemotherapies. In this concern, an antiestrogen drug skeleton was found to induce cytotoxicity towards breast-cancer cells that are resistant to the common antiestrogen drug. The efficiency of this drug has been related to the proton-coupled electron transfer observed in the presence of pyridine as a base [41].

**Conclusion**

It can be concluded that the pronounced results may encourage a country-wide project for not only collecting and getting rid of water Hyacinth but also making a pharma-economic value in Egypt. Harvesting water hyacinth, not only clean the drinking water from its deleterious effect but also would be used for the production of pharmaceutical remedies. In addition, they have scavenging antioxidant properties against the reactive oxygen species. The present study reveals the high potency of crude extracts as antioxidants and potential anticancer in comparison to their separated fractions. This crude extract represents favorable economic and industrial value for the production of commercial product. The pure compound separately or combined with other compound(s) or/and crude extract could be used as natural antioxidant and anticancer formulas.
Competing interests
The authors declare that they have no competing interests.

Authors’ contributions
Conceived and designed the experiments: ABE SMMS HAE. Performed the experiments: ABE SMMS EAH. Performed data analysis: SMMS. Wrote the paper: ABE SMMS HAE. Revised the manuscript: SMMS EAH.

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Author details
1. Department of Biochemistry, Faculty of Agriculture, Cairo University, 12613 Giza, Egypt.
2. Department of Botany and Microbiology, Faculty of Science, Cairo University, 12613 Giza, Egypt.
3. Department of Cell Biology, National Research Center (NRC), Dokki, Giza, Egypt.

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