Trace and Toxic Elements in the Soils of Different Ecosystems of Plateau Areas of the South Urals

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Abstract

The content and distribution of toxic elements in the rocks and soils of the Belebey plateau highlands, Ufa plateau and Zilair plateau were studied. The content of elements is determined by both natural and man-made factors. Natural accumulation is mainly related to the chemical composition of parent rocks. The parent rocks in Ufa plateau and Belebeyev plateau highlands are sedimentary with a big quantity of carbonates, and for the soils of the Zilair plateau – volcanic rocks. Man-made accumulation is characterized by a regression-accumulative type of distribution, which is manifested in an increased content in the humus-accumulative horizons and a sharp decrease in the lower ones. The accumulation of elements is associated with the extraction and processing of minerals, industrial emissions from enterprises, local wind and water transport of dust particles and aerosols from ash and slag dumps, and underground waste storage facilities. The amount of toxic elements in soils can also be determined by their belonging to various ecosystems (forest, arable land, meadow). The total chemical indicator in the research areas varies from strong to weak, and environmental conditions worsen in the areas: Belebey plateau highlands – Ufa plateau – Zilair plateau.

Keywords: soils contamination, hazard class, parent rock, soil

Introduction

The study of chemical elements, including toxic ones, is the primary task of modern soil science in conditions of increasing anthropogenic impact on the environment. For various reasons, natural environments are overloaded with compounds of heavy metals, and some of them constitute a very dangerous group of substances. Therefore, it is important to quantify these flows and study their potential impact on natural ecosystems [1, 2]. Typically, 14 elements are considered: mercury, lead, cadmium, arsenic, antimony, tin, zinc, aluminum, beryllium, iron, copper, barium, chromium, thallium. Heavy metals include more than 40 metals of the periodic system of Mendeleev with an atomic mass of more than 50 atomic units: V, Cr, Mn, Fe, Co, Ni, Cu, Zn, Mo, Cd, Sn, Hg, Pb, Bi, etc. At the same time,
not all of these elements are poisonous, some of them
are necessary for the functioning of humans, animals,
and plants [3-6].

Many chemical elements, especially microelements,
have become absolutely necessary for organisms (iodine,
zinc, copper, manganese, molybdenum, boron, etc.).
The elements carry out various catalytic and regulatory
functions of metabolic processes — absorption,
transport, oxidation-reduction, biosynthesis of organic
compounds, transmission of genetic information. Thus,
all the organisms, even if they are fully provided with
macrobiophiles (carbon, oxygen, hydrogen, nitrogen,
potassium, phosphorus, calcium), cannot develop
normally without the abovementioned elements. At
the same time, if at abnormally high concentrations,
they are inhibitory and toxic to plants, animals, and
humans [7-9]. An increase in the content of elements
(copper, zinc, chromium, molybdenum, manganese)
is accompanied by a positive biological response, then
the maximum level comes and then it drops to negative
values, i.e. the biological response of the body becomes
negative, and the element becomes toxic.

The founder of biogeochemistry, academician
V.I. Vernadsky [10] wrote: “Living matter covers
and regulates all or almost all chemical elements in
the biosphere. They are all needed for life and do not
enter the body by chance. There are no special life-like
elements. There are “dominant.” For example, scientists
Bertini and Gray [11] found about 10 tungsten-
containing enzymes in thermophilic microorganisms.
At the same time, tungsten is an element with a large
atomic number (74), although, as a rule, elements with
an atomic number of not more than 35 are used in
biological systems. The number of biological elements is
expanding today, including arsenic, bromine, strontium,
cadmium, barium, tungsten, tin. In this regard, the
content of the element, the form of its compounds and
the relationship with other elements are of decisive
importance.

Plants also have the ability to inactivate excess
amounts of heavy metals in tissues, which is determined
largely by their species characteristics [12-14]. For
each element, four levels of concentrations have to be
distinguished, in addition to the form of the compounds
and the relationship with the others: “element
deficiency”, when the body suffers from a deficiency,
“optimal content”, which contributes to a good state of
the body, “permissive concentrations”, when the body’s
depression is only beginning to appear, and “fatal
concentrations” for a given organism. In recent years,
the main attention of agrochemists, agroecologists and
soil scientists has been focused on studying the issues
of soil, natural waters and plants pollution, caused
by emissions from industrial enterprises. Currently,
standards for the content of a number of substances
have not been established, and the level of toxicity of
elements largely depends on the genetic properties of
the soil (content of humus, nutrients and secondary
minerals, pH of the medium, redox conditions, etc.),
the stability of natural plants and crops. Many scientists
[15-17] consider it more appropriate to orientate on
background values. The concentrations of substances
exceeding their natural content by 5, 10 or more
times are considered toxic. The main disadvantage of
the environmental assessment by MAC (maximum
allowable concentration) is the fact that the addition
of the negative effects of several elements, each of
which is present in subcritical concentrations, is not
considered [18, 19]. Obviously, if the content of many
elements is increased, then their combined effect can
lead to environmental disaster.

It is important to know the toxicants content in
soils and the environment for several reasons. First,
it will allow soil scientists to predict the behavior of
pollutants in the soil, depending on its particle size
distribution and mineralogical composition, degree
of humus content, acid-base and redox conditions, and
other properties. Secondly, it will help to solve the
problems of remediation of contaminated soils. Thirdly,
the correct design of artificial geochemical barriers
allows the successful purification of contaminated soil
and groundwater before it enters rivers or lakes. This
requires knowledge of the reactivity of the reagents
when fixing one or another pollutant to the barrier [20].

In the Republic of Bashkortostan, the content
of alkaline, alkaline-soil, rare-soil, and radioactive
elements in the soils of meadow, forest, and
agroecosystems has been fairly well studied. The
studies conducted in the soils of the Republic showed
the presence of anthropogenic and natural accumulation
of chemical elements in soils [21-24]. In the Republic,
there are different conditions for the formation of the
Earth crust, its mosaic, altitudinal and latitudinal
zonal and the related features of climate, soil and
vegetation. In this regard, a significant role in the
accumulation and migration of elements belongs to the
plateau-like highlands of the Southern Urals.

The aim of the work is a geochemical assessment
of the elemental composition of soils on plateau-like
highlands of the Southern Urals.

The research tasks included determining the content
of toxic elements in the soils of plateau-like highlands
of the Southern Urals, identifying differences in their
accumulation and distribution in the profile depending
on the geographical location.

The objects of research were the soils formed
on plateau-like hills: the Ufa plateau (300-520 m),
Belebey plateau highland (200-450 m), Zilair plateau
(300-600 m).

Material and Methods

The studies were conducted from 1999 to 2018
on permanent stationary sites in forest, meadow and
agroecosystems. A total of 22 full-profile soil sections
were laid, including on the Ufa plateau – 7 sections
in 4 stationary sections (Bayki, Maginsk, Abyzovo,
The geochemical assessment of soils according to the content of elements and the degree of contamination was carried out in accordance with the scales developed by Asylbaev, Khabirov [26, 27] for the conditions of the Republic of Bashkortostan (hereinafter referred to as the “scale”). When compiling the scales, the background contents characteristic of these territories were used. In the present work, the average data on the minimum amounts of elements in the soils of the region were taken as the background content. The indicators of the corresponding elements were compared with them in order to detect high, medium low and very low (background) contents (Table 1).

The total chemical indicator was calculated as the sum of the concentration coefficients of substances according to the Saeta formula [28]. The evaluation technique is as follows, the role of individual elements within their toxicity class was determined in relation to the background indicators and expressed as a percentage. The total chemical index was calculated as the sum of the concentration coefficients of substances according to the formula: 

$$Zc = \sum_{i=1}^{n} \frac{KCi - (n - 1)}{KCi}$$

where $n$ is the number of elements analyzed; $KCi$ is the concentration coefficient of the $i$-th element, $KCi = \frac{Ci}{Cfi}$, $Ci$ – the actual content of the element, $Cfi$ – background content [28]. The background values were determined by the minimum value or the average of several minimum values for each research area. The obtained calculations of $Zc$ were compared with the available categories of total soil pollution according to Table 2.

### Results and Discussion

The Ufa Plateau is a flat highlands with a strongly dissected karst-erosion relief composed of hard limestones, dolomites, variegated marls, and in places sandy deposits. The parent rocks are mainly represented by limestone eluvial-deluvial clays. On the Ufa Plateau, the soils of stationary sites are represented by gray forest (Greyic Phaeozems Albic), peaty humus residual-carbonate soils (Histic Cryosols Reductaquic). The humus-carbonate soils are more confined to mossy fir forests, pine forests and spruce forests. In these soils, under the poorly decomposed peaty litter, an organic-mineral horizon lies, consisting of humus and small fragments of crushed stone of carbonate rocks, the fine earth is often leached from carbonates, and boiling of...
10% hydrochloric acid is rapid only over fragments of crushed stone. The most common feature of soils formed on the Ufa plateau is the shortening of the soil profile, and the presence of carbonates in its lower part. In the process of soil development, with a decrease in the influence of the carbonate content of parent rocks, a podzolic process begins to appear in them. A feature of the region’s soils is also a relatively high humus content, which is due to the mineralization of a sufficiently large forest litter and plant debris in the continental climate and the close occurrence of calcareous bedrock, the composition and properties of which contribute to the neutralization of acid decomposition products and the fixation of humus in the form of calcium humates. These soils are saturated with bases, the reaction of the medium varies from slightly acidic to slightly alkaline.

On the Belebey plateau-like highlands, mainly black soils (Voronic Chernozems Pachic), and to a lesser extent gray (Greyic Phaeozems Albic) and dark-gray forest soils (Greyic Phaeozems Albic) are formed. The parent rocks are mainly deluvial carbonate clays, heavy and light loams and limestone eluvium. The chernozems are typical, for the most part, of medium power. Morphological features are the intense dark gray color of the humus horizon, good lumpy-granular structure, loose constitution, and an increased level of boiling from HCl. The humus content varies over a wide range from 4.7 to 12.8%, virgin soils are usually high humus, and arable soils are medium humus. Granulometric composition is heavy loamy, sometimes medium loamy. The reaction of the medium is close to neutral or slightly alkaline in the presence of free carbonates. The soils are saturated with bases, in which calcium predominates. These properties determine the high absorption capacity and buffering of typical chernozems, i.e. high environmental sustainability. Gray and dark gray forest soils are characterized by a lesser thickness of humus-accumulative horizon, gray and dark gray color, lumpy-fine-grained structure, and the presence of a pronounced densified illuvial horizon. The humus content in gray forest soil is 4.1%, and in dark gray - 7.9%, with depth, the humus content of both soils decreases sharply. The upper horizons are characterized by a slightly acid reaction of the medium, which becomes neutral towards the bottom of the profile. The formation of these soils on the eluvium of carbonate rocks led to a significant content of absorbed bases in the soil-absorbing complex.

On the Zilair Plateau, the soil cover is represented by dark gray forest (Greyic Phaeozems Albic), mountain dark gray forest (Leptosols Eutric) and meadow

| Elements | Levels of availability (pollution) of soil elements |
|----------|---------------------------------------------------|
|          | Permissible | Weak | Average | Strong | Very strong |
| Zn       | <50         | 51-100 | 101-150 | 151-200 | >200        |
| As       | <10.0       | 10.1-20.0 | 20.1-30.0 | 30.1-40.0 | >40         |
| Pb       | <10         | 10.1-15 | 15.1-20 | 20.1-25 | >25         |
| Cu       | <10         | 11-30 | 31-50 | 51-70 | >70         |
| Mo       | <0.5        | 0.51-1.0 | 1.1-1.50 | 1.51-2.0 | >2.0        |
| Co       | <5          | 5.1-10 | 10.1-20 | 20.1-30 | >30         |
| Ni       | <50         | 51-100 | 101-150 | 151-200 | >200        |
| Cr       | <50         | 51-100 | 101-150 | 151-200 | >200        |
| W        | <1.0        | 1.01-1.5 | 1.51-2.0 | 2.01-2.5 | >2.5        |
| V        | <100        | 101-200 | 201-300 | 301-400 | >400        |
| Sr       | <50         | 51-100 | 101-150 | 151-200 | >200        |
| Ba       | <100        | 101-500 | 501-1000 | 1001-1500 | >1500      |
| Th       | <4          | -    | 4-8   | 8-12   | >12         |
| U        | <5          | -    | 5-10 | 10-15 | >15         |

Zc – is the sum of the toxicants concentration coefficients in relation to the background indicators, calculated using the Saeta formula.
chernozem soils (Voronic Chernozems Pachic). These soils were formed on bedrock: dolomites, shales and sandstones. The thickness of the humus horizons of dark gray and gray forest soils A + AB is 38-54 cm, they are characterized by small reserves of humus, a low amount of absorbed bases and acidic reaction of the medium. Meadow-chernozem soils are characterized by a greater thickness of the humus horizon and a high content of exchange bases. In the humus horizon, the sum of the absorbed bases is 57-65 mEq per 100 g of soil, illuvial – about 47 mEq in parent rock – 40-54 mEq. The composition of the absorbed bases contains more than 2-3 times more calcium than magnesium. The humus content is average, the pH of the salt extract varies with depth from slightly acidic to slightly alkaline. Mountain dark gray forest soils are characterized by a lower thickness of the humus horizon, chastity, close to a neutral reaction of the environment. By granulometric composition they are heavy loamy. These soils are characterized by a relatively high humus content in the humus-accumulative horizon and a sharp decrease in its profile.

The geographical position, geomorphological and geological structure of the hills determined the differences in the accumulation of toxic elements in soils. The content of elements varies over a wide range from 0.1 to 103 mg/kg (Table 1). The accumulation of elements in soils is anthropogenic and natural. Nickel, strontium, chromium and uranium penetrate into the soils of plateau-like hills from parent rocks and in the soil profile their characteristics is similar (Fig. 2).

With a high content in parent and bedrocks, these elements on the one hand are carried away by trees, on the other hand, when toxicants with dusty air masses arrive, they settle on the leaves and crowns of trees and, after washing off with rainfall and leaf fall, enter the

![Fig. 2. Distribution of nickel along the soil profile.](image-url)
upper layers of the soil, and due to partial assimilation by trees they move into the root system and accumulate in the lower horizons of the soil.

The most common is the high content of copper, nickel, strontium and uranium in soils formed on sedimentary calcareous rocks of the Belebey Plateau Highlands and the Zilair Plateau, while in the Ufa Plateau this content is low. The soils of the Zilair plateau are characterized by a higher content of cobalt, chromium and tungsten, the Ufa plateau - arsenic, lead and thorium. This difference is explained by the fact that the soils of the Ufa plateau were formed on sedimentary rocks containing a large number of carbonates, and for the soils of the Zilair plateau volcanic carbonate-free rocks were parent rocks (Table 1).

The positive relationship was found between the content of toxic elements in the soil and in the rocks of the republic (Table 3). The functional dependence with the content of humus was not reliable. At the same time, theoretically, the strong positive anomalies of uranium detected in peat bogs are examples of the effective operation of natural organogenic geochemical barriers in the humid zone [29]. The influence of organic matter is especially significant: humic acids are able to form complexes with U (VI), facilitating the leaching of uranium from the parent rock, the fractions of organic matter available to microorganisms serve as a source of energy necessary for the recovery of U, humic acids also serve as an electronic shuttle for bacterial reduction of U. Biota due to enzymes acts as a catalyst for the recovery process U [30, 31].

A direct reliable dependence of the V content on silty particles was noted in the chernozems of the Central Chernozem region [32]. On the coast of Vigo in northwestern Spain, the V content in sediments correlates with the amount of silt fraction [33].

The content of other elements in parent rocks is lower than in soils, which is determined mainly by anthropogenic factors; they are characterized by a regression-accumulative type of distribution, which is manifested in increased accumulation in humus-accumulative horizons and a sharp decrease in the content in the lower ones. The accumulation of elements is associated with the extraction and processing of minerals, industrial emissions from industrial enterprises, local wind and water transport of dust particles and aerosols from ash and slag dumps, and underground waste storage facilities.

![Fig. 3. Distribution of lead over the soil profile of the Zilair plateau.](image-url)

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Table 3. Correlation dependence of the content of toxic elements in horizons A and AB on their concentration in the rock.

| Element   | y (A) from C (X) | y (AB) from X (C) |
|-----------|-----------------|------------------|
|           | r   | t<sub>r</sub> | Equation | r   | t<sub>r</sub> | Equation |
| Chromium  | 0.97 | 8.0   | \(y = 32.2 + 0.51x\) | 0.83 | 7.69 | \(y = 1.67 + 0.84x\) |
| Nickel    | 0.96 | 6.92  | \(y = 28.5 + 0.42x\) | 0.99 | 11.9 | \(y = 15.2 + 0.56x\) |
| Molybdenum| 0.77 | 2.41  | \(y = 0.62 + 0.16x\) | 0.83 | 2.96 | \(y = 0.55 + 0.27x\) |
| Strontium | 0.88 | 3.68  | \(y = 28.2 + 0.52x\) | 0.87 | 35.3 | \(y = 0.09 + 0.82x\) |
| Copper    | 0.88 | 3.74  | \(y = 0.88x - 2.88\) | 0.83 | 2.99 | \(y = 5.11 + 0.67x\) |
| Barium    | 0.91 | 4.47  | \(y = 305 + 29x\)   | 0.96 | 7.01 | \(y = 245 + 0.35x\) |
| Vanadium  | 0.91 | 4.3   | \(y = 16.1 + 0.66x\) | 0.99 | 12.62| \(y = 27.6 + 0.65x\) |
| Tungsten  | 0.69 | 1.89  | \(y = 0.43 + 0.32x\) | 0.95 | 6.39 | \(y = 0.27 + 0.45x\) |

\(r\) – is the correlation coefficient at 95% probability, \(t_\text{r} \) – Student’s criterion (2.45)
them are clay minerals, especially in small humus soils and sediments. Lead is concentrated in soils and sediments of heavy particle size distribution [34, 35]. The profile distribution of the lead content indicates its surface intake and in natural ecosystems the direction of movement is characteristic of the type of gray forest soils, and on the arable land the process of lead leaching in the illuvial horizon and accumulation in the parent rock are pronounced. A similar accumulation of cobalt and molybdenum in the upper horizons and a decrease in the lower horizons is characteristic of the arable land of the Zilair plateau and the Belebey plateau highlands. Typically, in soils involved in agricultural production, the gross lead content reaches 20 mg/kg of soil at a maximum permissible concentration of 100 mg/kg. It is established that liming of the soil and an increase in the response of the medium significantly reduce the lead content in plants. Lead in carbonate soils weakly enters plants, despite its increased content in soil solution [36]. Such a dependence of the intake

| Elements | Content limits | Belebey plateau-like highlands, n = 8* | Ufa plateau, n = 7* | Zilair plateau, n = 7* |
|----------|---------------|----------------------------------------|-------------------|----------------------|
|          |               | Rocks | Soils | Level of contamination | Rocks | Soils | Level of contamination | Rocks | Soils | Level of contamination |
| Zn       | max**         | 59.8  | 78.6  | Weak                  | 94    | 61.5  | Weak                 | 118.5 | 101.2 | Average               |
|          | min**         | 3.5   | 2.3   |                       | 29    | 26.7  |                      | 30.3  | 49.7   |                       |
| As       | max**         | 8.2   | 8.8   | Permissible           | 40    | 31.5  | Strong               | 11.8  | 9.7    | Permissible           |
|          | min**         | 2.1   | 1.3   |                       | 19    | 17.3  |                      | 1.5   | 4.6    |                       |
| Pb       | max**         | 10.6  | 15.9  | Average               | 26    | 32    | Very strong          | 18.9  | 18.2   | Average               |
|          | min**         | 5.1   | 6.5   |                       | 1.4   | 1.1   |                      | 4.5   | 7.7    |                       |
| Cu       | max**         | 32    | 63.6  | Strong                | 30    | 15.5  | Weak                 | 92.5  | 80.4   | Very strong           |
|          | min**         | 15.3  | 19.9  |                       | 0.1   | 0.7   |                      | 2.4   | 1.1    |                       |
| Mo       | max**         | 0.51  | 0.7   | Weak                  | 1     | 0.7   |                      | 2.4   | 1.1    |                       |
|          | min**         | 0.1   | 0.1   |                       | 0.1   | 0.3   |                      | 0.3   | 0.5    |                       |
| Co       | max**         | 17.9  | 19.3  | Average               | 13    | 12    | Average              | 35.1  | 27     | Strong                |
|          | min**         | 6.6   | 6.5   |                       | 7.4   | 4.9   |                      | 4.1   | 8.3    |                       |
| Ni       | max**         | 191.5 | 169.6 | Strong                | 64    | 43    | Permissible          | 450   | 408.8  | Very strong           |
|          | min**         | 37.1  | 38.5  |                       | 36    | 28.5  |                      | 45    | 58     |                       |
| Cr       | max**         | 132.8 | 126.8 | Average               | 76    | 71.5  | Weak                 | 317.9 | 283.2  | Very strong           |
|          | min**         | 23.2  | 27.2  |                       | 34    | 33    |                      | 129   | 118    |                       |
| W        | max**         | 1.26  | 1.7   | Average               | 1.0   | 1.0   | Permissible          | 3.2   | 2.4    | Strong                |
|          | min**         | 0.1   | 0.1   |                       | 0.7   | 0.6   |                      | 0.7   | 0.9    |                       |
| V        | max**         | 68    | 103.2 | Weak                  | 200   | 152   | Weak                 | 162.4 | 133.8  | Weak                  |
|          | min**         | 31.7  | 55.1  |                       | 34    | 39.5  |                      | 72.6  | 86     |                       |
| Sr       | max**         | 415   | 187.3 | Strong                | 120   | 88    | Weak                 | 336.6 | 247.3  | Very strong           |
|          | min**         | 100.8 | 80.1  |                       | 82    | 64.3  |                      | 59.2  | 65.3   |                       |
| Ba       | max**         | 634   | 630   | Average               | 206   | 218.5 | Weak                 | 413   | 412.7  | Weak                  |
|          | min**         | 29.5  | 195.8 |                       | 139   | 132.3 |                      | 63.6  | 95.4   |                       |
| Th       | max**         | 2.0   | 6.1   | Average               | 16    | 13.5  | Very strong          | 0.3   | 2.4    | Permissible           |
|          | min**         | 0.3   | 1.9   |                       | 9.3   | 7.8   |                      | 0.1   | 1.2    |                       |
| U        | max**         | 18.0  | 16.8  | Strong                | 3.0   | 2.7   | Permissible          | 19.2  | 16.2   | Very strong           |
|          | min**         | 2.0   | 2.3   |                       | 1.6   | 1.5   |                      | 8.0   | 10.3   |                       |

* n – is the number of sections, ** max and min – are the highest and lowest average element content for the soil profile.
of lead into plants on the carbonate content in the soil and the pH value follows, first of all, from a decrease in the thermodynamic activity of free lead ions, that is, ions that are not associated with other ions. In alkaline soil solutions of carbonate alkaline soils, the molar fraction of the active concentration (activity) of lead ions does not exceed 0.13%, in aqueous extracts - 0.24%. Whereas in cadmium, the molar fraction of the active ion concentration is much higher and amounts to 4.0 and 11.2%, respectively [37] (Fig. 3).

The Saeta formula allowed to evaluate the degree of total soil contamination with toxic elements. The data obtained by us make it possible to assess the ecological situation of the geomorphological regions of the Republic of Bashkortostan by the degree of total soil pollution with toxic elements. According to the content of elements and their total chemical index, the regions vary significantly. The total chemical indicator in the research areas varies from strong to weak, and environmental conditions worsen in the following areas: Belbebay Plateau Highlands - Ufa Plateau - Zilair Plateau (Table 4 and 5).

In the Ufa Plateau, the largest proportion among the elements is lead (38.9%), where the excess of the background is 15 times (max-32 mg/kg). The average lead content in the soils of the world varies from 3 to 180 mg/kg, in China – from 17 to 280 mg/kg [38]. The natural content of lead in soils is inherited from parent rocks, however, due to large-scale aerogenic pollution in the upper layers of the soil, it can accumulate due to this. In acidic igneous rocks and clay sediments, lead is in the range of 10-40 mg/kg, in ultrabasic and calcareous rocks, 0.1-1.0 mg/kg. In the Ufa Plateau, the content in carbonate rocks was 26 mg/kg. During anthropogenic pollution, the highest element concentration is usually found in the humus-accumulative layer of soil [39]. According to the data of G.D. Chimitdorzhiev et al. [40], in the meadow chernozem permafrost soils and in the parent rock, the lead content averaged 12.5 mg/kg, increasing in the upper horizon to 22.0 mg/kg, which is much more than its clarke (13 mg/kg).

In technologically contaminated podzolic soils of the Kola Peninsula, in the upper horizons, on average, 46 mg/kg of zinc with fluctuations from 12 to 198 mg/kg is contained, and in the C-horizons from 3.7 to 348, on average, 20.9 mg/kg [42].

In the Belebey plateau-like highlands, the main pollutant is nickel, which accounts for 15.1% of the total amount and a 4-fold excess of the background value. For other elements, the excess of background values is from 1.2 to 2 units. The nickel content in the rocks of the Zilair
Plateau is 450 mg/kg, in the soil 408.8 mg/kg, which is also much higher than the MAC standard (85 mg/kg) [44]. On average, the soils of the Republic of Belarus contain 20.0 mg/kg Ni, the average Ni content in peat soil is 4 mg/kg [45]. The analysis of data on the Ni content in sod-podzolic soils showed that its content in all soil samples is significantly lower than the regional clarke [46].

The highest concentrations of nickel, as a rule, are observed in clayey and loamy soils, in soils formed on the main and volcanic rocks and rich in organic matter. The distribution of Ni in the soil profile is determined by the content of organic matter, amorphous oxides and the amount of clay fraction. On the coast of Vigo Island in northwestern Spain, the average nickel content in clay sediments is three times higher than in sandy ones [47]. A direct reliable dependence of the nickel content on the amount of silty particles was noted in the gray forest soils of the Central Black Earth Region [48]. Nickel also manifests itself as a siderophile. On the coast of Vigo Island in Spain, the amount of Ni in the sediments reliably correlates with the total iron content: \( r = 0.63 \), which indicates the affinity of nickel with iron (hydr)oxides [35].

**Conclusions**

The geographical position and geomorphological structure of the highlands have a number of common features and differences that determine the nature of the accumulation of elements. The most common is the increased content of elements in the soils and rocks of these elevations and is determined by both natural and man-made factors. The natural accumulation of cobalt, strontium, vanadium, tungsten, arsenic and nickel is mainly due to soil-forming processes and their content in soil-forming, acidic and basic rocks.

The accumulation of lead, zinc, copper, chromium, molybdenum and barium is determined mainly by man-made factors; they are characterized by a regression-accumulative type of distribution, which manifests itself in increased accumulation in humus-accumulative horizons and a sharp decrease in the content in the lower ones. The accumulation of elements is associated with the extraction and processing of minerals, industrial emissions from industrial enterprises, local wind and water transport of dust particles and aerosols from ash and slag dumps, and underground waste storage facilities.

Among the elements in all soils, the main pollutants are copper, nickel, strontium and uranium, and an acceptable level of content is characteristic of molybdenum, zinc and vanadium. Under conditions of predominantly natural pollution factors, strong and very strong categories were identified in the Zilair plateau and are caused by the accumulation of copper, cobalt, nickel, chromium, tungsten, strontium and uranium; Belebey plateau-like highlands - copper, nickel, strontium and uranium; Ufa plateau - arsenic, lead and thorium.

The amount of toxic elements in soils is largely determined by their belonging to various ecosystems. In the soils under the forest, the accumulation of many elements, especially strontium, tungsten, copper, chromium, cobalt, nickel, molybdenum, occurs in different ways. In arable land soils, high concentrations of cobalt and molybdenum are mainly due to their introduction with fertilizers, ameliorants and pesticides. The obtained data can become the basis for long-term and retrospective monitoring of the geochemical state of soils. According to the results of the study, environmental soil passports for individual geomorphological regions can be developed, which is also advisable to take into account when determining their market value. According to the environmental assessment and taking into account the reserves of chemical elements, areas and soils have been established which have the purity level allowing to produce environmentally friendly agricultural products.

**Conflict of Interest**

The authors declare no conflict of interest.

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