Review on fluorinated nucleoside/non-nucleoside FDA-approved antiviral drugs

Magda M. F. Ismail and Mohammed Salah Ayoup

FDA-approved antiviral agents represent an important class that has attracted attention in recent years to combat current and future threats of viral pandemics. Fluorine ameliorates the electronic, lipophilic and steric problems of drugs. Additionally, fluorine can prolong drug activity and improve metabolic stability, thereby, modifying their pharmacodynamic and pharmacokinetic character. Herein, we summarized the fluorinated FDA-approved antiviral agents, dealing with biological aspects, mechanisms of action, and synthetic pathways.

1. Introduction

Viruses are amongst the dominant pathogenic organisms that cause a lot of severe diseases. Advancing approaches for the treatment of viral infections focuses on either the viruses or the infected host cell. Some antiviral agents act directly as inhibitors of viral attachment, uncoating, entry, or fusion, and some inhibit RNA polymerase, protease, integrase, nucleoside, or nonnucleoside reverse transcriptase. During the 2010s, the protease viral DNA polymerase and integrase inhibitors were indexed among the top 200 marketed drugs. To date, there is no effectual antiviral agent for numerous viral diseases. Nevertheless, a pair of drugs has been discovered for herpes viruses, many drugs for influenza and a number of therapeutics for hepatitis C infection and human immunodeficiency virus type 1 (HIV-1).

Antiviral drugs are primarily transformed to their triphosphate then inhibit viral DNA synthesis. Mechanisms of famous antiviral drugs have been analyzed revealing their ability to improve the body’s resistance to a virus, repress adsorption or diffusion of the virus into the cell and denature viral protein. The inhibition of nucleic acid synthesis is another mechanism for antimetabolites.

The modifications brought to the ribose or 2'-deoxyribose moiety of a nucleoside comprise changes in the sugar moiety, oxygen replacement with an isostere, ring size changes, or replacement with an open-chain moiety. These modifications may lead to astonishing discrepancies in toxicity and biological assets.

It is well known that the high electronegativity of fluorine results in the polarization of adjacent bonds resulting in acidity enhancement and a decrease in the basicity of the substituents in close proximity. The electronegativity gives rise to a more polarizable C–F bond with increased electrostatic character. Conventionally, C–F bonds can bind with their receptors by either dipole or electrostatic interaction. The pKₐ is increased, which can affect the potency, selectivity, and toxicity. The hydroxyl group is always replaced by fluorine, which is characterized by the same polarity and ability to form a hydrogen bond. The introduction of a fluorine atom or trifluoromethyl group (–CF₃) changes both the pharmacodynamics and pharmacokinetics (absorption, distribution, metabolism, and excretion) of the molecule. Fluorine is a metabolic blocker; the blocking occurs when the C–H bond is replaced by a C–F bond which has a high impact on the biological half-life and bioavailability enhancement. Fluorine is small compared to the hydrogen atom, therefore, replacing the hydrogen atom with fluorine can considerably increase the binding affinity of a drug with receptors via the formation of noncovalent bonds. On the other hand, cell penetration is increased due to less plasma protein binding and a higher free fraction of the drug. These outcomes lead to a greatly improved therapeutic activity.

Among various antiviral drugs presently used, this review will address the mechanisms of action, synthetic pathways, and therapeutic uses of the fluorinated U.S. Food and Drug Administration (FDA)-approved antiviral agents over the last four decades. Furthermore, the biological value of the fluorine substituent will be clarified for the antiviral activity against HIV type-1, COVID-19, influenza, smallpox, acute asthma, and hepatitis C.

2. Treatment of smallpox

Tecovirimat (ST-246) has been endorsed in the United States and approved in 2018 as a therapy for smallpox, monkeypox, and orthopoxvirus diseases for which it showed potent antiviral...
activity. This drug blocks the viral escape from an infected cell, burdening the dissemination of the virus within the body. The para-trifluoromethyl group has an important role in the optimal metabolic stability and pharmacokinetic profile of this drug. Teicovirimat (7) was synthesized in 2014 by Dai using versatile methods, starting from the reaction of maleic anhydride (1) with monoBoc hydrazine to afford the imide derivative 2. Boc deprotection of 2 using 4 M HCl afforded N-amino maleimide hydrochloride 3 which can be acylated by p-trifluoromethyl benzoyl chloride 4 to give the hydrazide derivative 5. Then, cyclo-addition of 5 was achieved with cycloheptatriene in xylene to produce teicovirimat, 7, in excellent yield as shown in Scheme 1.33

Another modified protocol was employed to prepare tecovirimat starting from the cyclo-addition reaction of maleic anhydride, 1, with cycloheptatriene, 6, to yield the adduct 8, which after reaction with BocNHNH₂ provided the fused imide derivative, 9. Deprotection of 9 in 4 M HCl gave rise to the free hydrazide derivative, 10, which is finally acylated by p-trifluoromethylbenzoyl chloride, 4, to afford 7 (Scheme 2).34

3. Treatment of acute asthma

Pleconaril (Picovir) was originally manufactured by Schering-Plough, who made an agreement with ViroPharma in 2004 to license intranasal pleconaril for the treatment of the common cold in the US and Canada. Pleconaril is active against viruses in the picornavirus group; it is accommodated in a hydrophobic pocket of the virion, which leads to its rigidification and viral capsid compression in the capsid structural protein. Consequently, it prevents the viral attachment and uncoating process in the host cell, a crucial step in the life cycle of the virus.15 Pleconaril is employed for the prevention of asthma exacerbation and currently, its intranasal spray formulation is in clinical stages for treatment of rhinovirus infections, while the oral medication is only formerly used due to reported aftereffects. A remarkable enhancement of its activity was observed due to the combination of two factors: the bulky trifluoromethyl group which interferes with the enzyme binding, and its effect as a metabolic blocker, especially against cytochrome P450 enzymes to protect the oxadiazole ring from hydroxylation.35

The synthesis of pleconaril (18) by Diana et al. in 1995 is shown in Scheme 3, starting from the reaction of dimethyl oxazolale, 11, with oxirane (12) via carbon–carbon bond formation using BuLi to afford the primary alcohol, 13, which provided the corresponding chloro derivative, 14, upon treatment with thionyl chloride. Then, Williamson synthesis was carried out via reaction of 14 with the phenol 15 via S_N2 reaction, to yield the ether 16 which upon addition of hydroxylamine to the nitrile functionality gave the corresponding amidoxime, 17. Subsequently, the latter was cyclized using TFAA to afford pleconaril (18).36–39

4. Treatment of HIV type-1

When the immune system is attacked by HIV without treatment, severe acquired immunodeficiency syndrome (AIDS) will result, and unfortunately, there is no effective cure yet.1,40 The presently accepted anti-HIV agents can be categorized as inhibitors of reverse transcriptase, integrase, protease, fusion, and co-receptor. Combination therapy using antiretroviral agents greatly suppresses viral growth and replenishes the immune system.41,42 Development of drug resistance, in addition to their toxicity, limits the use of these drugs.8,43,44

Maraviroc (MVC) has commercial names such as Selzentry and Celsentri and was approved in 2007; it acts as a selective HIV entry and fusion inhibitor for C–C chemokine receptor type-5 (CCR5), and is accepted for the treatment of advanced patients infected with CCR5-tropic HIV-1. Much effort have been applied to decrease the affinity of these agents towards

Scheme 1 Synthesis of teicovirimat (7) from maleic anhydride.
human receptors closely correlated with CCR5 e.g. the potassium channel HERG, responsible for the cardiotoxicity that appears with this type of medicament.\textsuperscript{45} Combination therapy of MVC with other medications is advised to cure a certain type of human immunodeficiency virus (HIV) infection. The complete synthesis of maraviroc \textsuperscript{33} was described in a US patent\textsuperscript{46} by Igor et al. in 2014 via the formation of tropinone, \textsuperscript{22}, by Robinson synthesis using benzylamine, 2-oxomalonic acid, and a furan derivative. Condensation of \textsuperscript{22} with hydroxylamine affords the corresponding oxime \textsuperscript{23}, which upon subsequent reduction via reflux with sodium in penthanol afforded the exoamine \textsuperscript{24}, which reacted with \textsuperscript{25} to yield the corresponding amide, \textsuperscript{26}. Cyclocondensation of \textsuperscript{26} with acid hydrazide \textsuperscript{27} afforded the 1,2,4-triazole derivative, \textsuperscript{28}, which is then deprotected to \textsuperscript{29} under transfer-hydrogenolysis conditions. Next, the addition of the tropane-triazole adduct \textsuperscript{29} to the aldehyde \textsuperscript{30} occurred followed by deprotection of the Boc group to yield \textsuperscript{31}. Peptide coupling of \textsuperscript{31} with the difluorinated acid, \textsuperscript{32}, afforded the target antiviral drug maraviroc (\textsuperscript{33}) (Scheme 4).\textsuperscript{47,48}

Integration of the viral DNA into the host genome required an essential enzyme, HIV integrase, which is a well-executed target in combating HIV where there is no human homolog. Integrase inhibitors are composed of two distinctive constituents. The first is a diketo acid moiety, which is capable of forming covalent bonds with DNA, with the aid of integrase enzyme; then, the sealing of the viral DNA into the chromosome occurs by cellular repair activities. The second moiety is the benzyl group with lipophilic nature which permits binding to an important hydrophobic pocket of integrase.\textsuperscript{49} The fluorine attached to the benzyl moiety results in a higher affinity between the drug and the protein via binding to numerous nonpolar residues within the hydrophobic pocket of the integrase enzyme. In the first generation of HIV integrase inhibitors, the para-fluorophenyl is in a favored position e.g. in raltegravir (RAL), or the ortho-fluorophenyl e.g. in elvitegravir,
while second-generation inhibitors like dolutegravir have the ortho/para bis-fluoro phenyl as an essential moiety.

Raltegravir is a first generation HIV integrase strand transfer inhibitor (INSTI), FDA-approved in 2007 and sold under the trade name Isentress. It inhibits the role of HIV integrase, required for viral reproduction. The para-fluoro group is capable of binding with the former viral nucleotide of the enzyme. RAL can be utilized as a prophylactic or as combined therapy with other drugs to treat HIV/AIDS. It is worth mentioning that RAL and other integrase strand transfer inhibitors are well-tolerated antiretroviral drugs (ARVs).38 Raltegravir 44 was synthesized in 2013 by Gurjar et al., starting from acylation of 2-amino-2-methyl propanenitrile, 35, by oxadiazole acid chloride, 34, in the presence of NMM, 36, to afford the nitrile, 37. Then, the addition of hydroxylamine to compound 37 yielded the corresponding amidoxime, 38, which upon cyclization with dimethylacetylene-dicarboxylate, 39, provided the dihydropyrimidine ester derivative, 40. Amidation of the latter by p-fluorobenzylamine, 41, gave the norraltegravir, 42, which undergoes N-methylation using trimethylsulphoxide iodide, 43, to afford raltegravir (44) (Scheme 5).39

Elvitegravir (EVG) 56 was first approved in 2012 and is utilized for HIV therapy; it acts via inhibition of the HIV-1 integrase enzyme and it also inhibits the integrase enzyme of HIV-2 to a lesser degree.50 Phase II clinical trials showed that patients taking elvitegravir once a day enhanced by ritonavir (protease inhibitor, PI) had a more significant decline in viral burden when compared to patients receiving a ritonavir-boosted protease inhibitor after the same time period. Clearly, ortho-fluoro in EVG has a favorable role where it contributes to van der Waals interactions with glutamate residues in the enzymatic pocket. The synthetic procedure for EVG was achieved by Stanislav et al. in 2014 via the reaction of difluorobenzoic acid 45 with thionyl chloride to give the corresponding acid chloride 46. Treatment of the latter with the ester 47 allowed the transformation to benzoyl acrylate 48; then, a nucleophilic substitution reaction with S-(+)-valinol 49 provided the enamine 50, which was then condensed to quinolin-4-one 51. Activation of the primary alcohol of 51 was
carried out to produce the carbonate ester 52 using methyl chloroformate; then Negishi cross-coupling with an organo-zinc reagent 53 in the presence of a palladium catalyst provided the benzyl quinolone 54. This was followed by alkaline deprotection to yield the corresponding free primary alcohol 55, which was subjected to reaction with sodium methoxide to provide elvitegravir (56) (Scheme 6).

Dolutegravir (DTG) 69 was FDA approved in 2013 and is marketed with the commercial name Tivicay; it is an antiretroviral therapy used, combined with other drugs, to treat HIV/AIDS. A combination of dolutegravir/abacavir/lamivudine is also available. Notably, DTG shows longer duration of action than raltegravir; this may be attributed to the amide linker attaching the diketo acid to the benzyl group which allows the difluorobenzyl moiety to interact more strongly with the lipophilic pocket of the integrase enzyme. 69 Since 2019, the WHO has advocated DTG as the first- and second-line treatment for everyone with HIV. The synthesis of DTG proceeds as mentioned in the US patent by Sumino et al. in 2012 and Maras et al. in 2016, starting from the condensation of ethyl chloroacetooacetate 57 with the acetal 58 to afford the enaminone 59 which was subsequently cyclized with chloro-oxalate 60 to give 4-pyrene derivative 61. This was transformed to the corresponding 4-pyridone 63 via reaction with acetalamine, 62. Then, acidic hydrolysis of 63 gave the corresponding aldehyde 64; treatment of the latter with butanolamine 65 afforded the fused tricyclic ester 66. Acidic hydrolysis of 66 gave 67 which underwent coupling with difluoro-benzylamine 68 to provide the target antiviral 69 (Scheme 7). 55,56

Tipranavir (TPV) 78 was approved by the FDA in 2005 for pediatric use to treat HIV infection. Only combination therapy of tipranavir/ritonavir is allowed to be administered. Among protease inhibitors, tipranavir is capable of inhibiting viral replication which is resistant to other medications. Two mechanisms may elucidate the potency of tipranavir against C-resistant HIV. First, it may bind to protease’s active site with a smaller number of hydrogen bonds than peptidic PIs, leading to more flexibility to tune amino acid substitutions at the active site. 57 Second, the presence of a trifluoromethyl pyridyl-sulfonamide moiety enhances the pharmacodynamic and pharmacokinetic profile. 58 Some side effects of tipranavir include intracranial hemorrhage, hepatitis, hyperglycemia and diabetes mellitus, in addition to an increase in total cholesterol and triglycerides.

Tipranavir (78) was prepared by Romines et al. in 1998 via an efficient and short enantio-selective synthesis starting with allylic alkylation of 70 with methyl sodiomalonate using molybdenum complex Mo(CO)3(C5H8) and the ligand 71 as a catalyst. The key compound 73 was obtained in an excellent yield from the malonate derivative 72. Next, aldol coupling of
the ester derivative 73 with the aldehyde 74, followed by Dess–Martin oxidation afforded β-ketoester 75 (1 : 1 mixture). The pyrone 76 was obtained after cleavage of the PMB group, and subsequent reduction of the NO₂ functionality followed by treatment of the resulting amine with trifluoromethyl pyridine sulfonfyl chloride 77 to provide tipranavir (78) (Scheme 8).⁵⁹,⁶⁰

Viral infections are classified into three basic classes; the first one encompasses lethal viruses such as HIV, hepatitis B, and C viruses. The second class encompasses acute infections like influenza and the third class includes viral infections e.g. the common cold which is nonlethal.⁶¹,⁶² Fluorinated nucleoside reverse transcriptase inhibitors (NRTI) such as emtricitabine (FTC) and racivir (RCV) are a notable category of FDA-approved medications for the treatment of HIV type-1.

Emtricitabine (88) is a 5-fluoro deoxycytidine approved in 2003 and is in the index of important drugs assembled by the WHO.⁶³ It is therapeutically employed for treatment of HIV infection in both adults and children via reverse transcriptase inhibition.⁶⁴,⁶⁵ The FDA in 2006 accepted a three-drug combination (Atripla), emtricitabine, tenofovir and efavirenz. Additionally, the WHO registered two other combinations: emtricitabine/tenofovir and emtricitabine/efavirenz as indispensible drugs. Emtricitabine consists of a cytosine base with a fluorine atom at the C-5 position and an uncommon oxathiolane ring as a sugar moiety. It was realized that the activated form of FTC (triphosphate metabolite) was integrated faster into growing viral DNA than that of lamivudine. Furthermore, the introduction of fluorine improves its bio-accessibility parameters and duration compared to lamivudine.⁶⁶ A stereo-selective synthetic route of emtricitabine (88) was accomplished by Mansour et al. in 1997, starting from the formation of trans-oxathiolane derivative 81 from the reaction of 2,5-dihydroxy-1,4-dithiane (79) and glyoxylic acid (80). Acetylation of 81 afforded 82 followed by esterification with a chiral auxiliary 1-menthol (83) which led to a mixture of diastereoisomers 84 and 85. The desired diastereoisomer 84 was isolated by fractional crystallization and directly coupled with silylated 5-fluorocytosine (86), to provide the N-nucleoside derivative 87. Finally, reductive removal of the chiral auxiliary from 87 gives emtricitabine 88.

![Scheme 6](image_url)

Scheme 6 Synthesis of elvitegravir (56) from difluoro-iodobenzoic acid 45.
Scheme 7  Synthesis of dolutegravir 69 from ethyl chloroacetoacetate 57.

Scheme 8  Synthesis of tipranavir (78) using molybdenum-catalyzed allylic alkylation.
represented with inversion of configuration in position 5 of oxathiolane\textsuperscript{67–70} (Scheme 9).

Racivir is an enantiomer of emtricitabine. It is a widely used nucleoside reverse transcriptase inhibitor advanced by Pharmasset Company for HIV treatment and is currently in phase 2b clinical trials; it blocks a further increase in the amount of HIV virus in the body.\textsuperscript{71} Racivir is a racemic mixture of the two \(\beta\)-enantiomers of emtricitabine (88), and a neat

Scheme 9  Synthesis of emtricitabine (88) from trans oxathiolane derivative 81.

Scheme 10  Synthesis of racivir from 2,4-dichloro-5-fluoropyrimidine (89).
synthetic method has been patented by Glaxo Wellcome Inc. Its synthesis was achieved by reaction of 2,4-dichloro-5-fluoropyrimidine (89) with NaOEt to yield 90; which is then reacted with the anion of 2,2-dimethoxyethanol (91) to produce the pyrimidine derivative 92. Deprotection of the latter gave the corresponding aldehyde 93. Then, the formation of 1,3-oxathiolane 94 occurred via reaction of 93 and 2,5-dihydroxy-1,4-dithiane 79. 1,3-Oxathiolane 94 was later acetylated to give 95 which upon treatment with TMSOTf afforded racemic 96. Eventually, the reaction of 96 with ammonia led to the racemic mixture 97 (racivir) (Scheme 10).

Non-nucleoside reverse transcriptase inhibitors (NNRTIs) are dissimilar to NRTIs where metabolic activation is not required. Alternatively, they act as noncompetitive blockers targeting the nearby allosteric site of HIV-1 RT, which leads to conformational changes and impairment of the viral replication. Udit et al. reported in 1999 the synthetic pathway of efavirenz 108. The complete synthesis starts from the acylation of p-chloroaniline 98 by t-buty carbonyl chloride to afford the amide 99, followed by acylation of the aromatic ring using BuLi and trifluoroacetate to give the trifluoromethyl ketone derivative 100. Acidic hydrolysis of 100 afforded the O-trifluoroacetyl aniline derivative 101, and then N-alkylation of 101 using p-methoxybenzyl chloride (102) provided the corresponding N-PMB aniline derivative 103. Asymmetric addition of the lithium salt of cyclopropyl acetylide, 105, to the ketone group of 103 in the presence of the chiral catalyst 104 afforded the pure tertiary alcohol 106 in excellent enantiomeric excess. Cyclization of 106 was achieved by treatment with phosgene gas to afford the protected N-PMB efavirenz 107 which was deprotected by a cerium nitrate and ammonium nitrate protocol to provide efavirenz 108 (Scheme 11).

5. Treatment of hepatitis C

Sofosbuvir (Sovaldi®, GS7977) 113 is an RNA-dependant RNA polymerase inhibitor approved by the FDA in 2013, manufactured by Gilead Sciences, Inc. It is a prodrug and has the advantage that the active phosphorylated nucleotide enhances both cell permeability and oral bioavailability. It is the nucleotide analog that interferes with the HCV lifecycle, curtailing viral replication. This prodrug is utilized along with ribavirin and sometimes another medication to treat certain types of chronic hepatitis C in adults. Sofosbuvir appears to have a high threshold to the development of resistance. Among sugar-fluorinated nucleosides, sofosbuvir is authorized by the FDA as a therapy for chronic HCV infections including for patients who suffer from hepatocellular...
carcinoma and others with HCV/HIV-1 co-infection. Its preparation was accomplished by Kaushik et al. in 2015 via the reaction of fluorinated deoxynucleoside 110 with the phosphonate 109 which contains a good leaving group (X) such as tosylate, camphorsulfonate, mesylate, trifluoroacetate, trifluorosulfonate, an aryloxy, or heteroaryl oxide substituted with at least one electron-withdrawing group to afford the key compound phosphonucleoside 111. Then, debenzoylation was carried out via reaction of the latter with aqueous acetic acid 70% to produce 112. Subsequently, hydrolysis in the last step was performed using an aqueous acid such as hydrochloric acid, hydrobromic acid, sulfuric acid, formic acid, acetic acid, fumaric acid, oxalic acid, or mixtures thereof to provide sofosbuvir 113 (Scheme 12).65

Ledipasvir was developed by Gilead Sciences. It inhibits an important hepatitis C virus protein (NS5A), which is incorporated into viral assembly, secretion, and replication.66 Interestingly, patients with genotypes 1a or 1b recover upon using interferon or ribavirin.67 Metabolism of sofosbuvir generates active uridine triphosphate which causes RNA chain termination, when integrated in RNA by NS5B polymerase. The addition of ribavirin to this combination will result in an effective therapy for chronic hepatitis C in both children and adults. Ledipasvir was prepared by Scott et al. in 2013 via acylation of 2-bromo-9,9-difluoro-7-ido-9H-fluorene (114) with 2-chloro-N-methoxy-N-methylacetamide (115) in THF with a catalytic amount of i-PrMgCl at −10 to 0 °C to afford the chloroacetyl-fluorene derivative 116. Then, the nucleophilic substitution of 116 was carried out with the potassium salt of the spiro compound 117 in acetone to produce the corresponding ester 118. Subsequently, reflux of the latter in methoxyethanol with ammonium acetate was performed to provide the imidazole derivative 119 via a cyclization mechanism. New C–C bond formation occurred via the treatment of 119 with boron compound 120 using a palladium acetate, Mephos, and potassium acetate protocol to produce 121. Then, Boc deprotection of 121 was achieved using HCl to give the tetrahydrochloride salt 122. Ultimately, bis-peptide coupling of 122 with (S)-2-(methoxycarbonylamino)-3-methylbutanoic acid (Mocvaline) (123) using an EDC-HCl/NMM/HOBt/DMF protocol afforded the target ledipasvir (124) (Scheme 13).88,89

6. Treatment of eye infections

Trifluridine 127 (trifluorothymidine or TFT) was FDA approved in 1980. It was sold under the trade name Viroptic by Glaxo Wellcome; now it is merged into GlaxoSmithKline as an ophthalmic drug. It is utilized for viral eye infections caused by the herpes simplex virus in particular, and does not affect bacteria or fungi.80 Furthermore, combined therapy of trifluridine and tipiracil is beneficial as an oral anticancer agent for colon or rectal cancers. Trifluridine is a thymidylate synthetase and DNA polymerase inhibitor. It is a suicide substrate that is converted to an irreversible inhibitor at the active site of the enzyme, preventing the conversion of deoxyuridine monophosphate (dUMP) into thymidine monophosphate (TMP), a crucial step in DNA biosynthesis. This will lead to apoptosis, and alter viral or cancer cell replication.81

The synthetic procedure of trifluridine, 127, was achieved by Heidelberger et al. in 1962 via a free radical reaction of uridine (125) with Togni’s reagent (126) in the presence of catalytic amounts of cuprous salt. Alternatively, trifluridine 127 could be prepared directly by reaction of 5-iodouridine (128) with CF₃Cu (that generated from the reaction in situ of CF₃I and copper powder)91,92 as shown in Scheme 14.

7. Treatment of influenza and corona viruses

Favipiravir (T-705) (134) was approved for medical use in Japan by Toyama Chemical (a subsidiary of Fujifilm) in 2014. Then, in

![Scheme 12](image-url)  
**Scheme 12**  
Synthesis of sofosbuvir (113) from fluorinated deoxynucleoside 110.
Scheme 13  Synthesis of ledipasvir (124) from 2-bromo-9,9-difluoro-7-iodo-9H-fluorene (114).

Scheme 14  Synthesis of trifluridine (127) from uridine.
2016, it became a generic drug and was sold under the brand name Avigan in 2019. T-705 has a selective and broad-spectrum antiviral effect towards the RNA-dependent RNA polymerase (RdRp) of RNA viruses. It is a fluorinated pyrazine carboxamide derivative, used to treat influenza in Japan. Concerning the oral formulation, the addition of a 6-fluoro group at the pyrazine ring enhanced both the biological potency and pharmacokinetic character of T-705 in comparison to other derivatives. In addition, patients of COVID-19 suffering from a mild-to-moderate illness treated with T-705 show viral clearance and clinical improvement.

The synthesis of favipiravir (134) was achieved by Furuta et al. in 2000 as shown in Scheme 15, starting from pyrazine-2-carboxylate derivative 129 which was transformed to its methoxy derivative 130 via diazotization in methanol. Next, 130 was subjected to Buchwald–Hartwig amidation to give 131; followed by amidation utilizing aq. ammonia to afford the corresponding amide 132. Diazotization of 132 was carried out in the presence of pyridine hydrofluoride (Olah’s reagent) to produce the fluorinated pyrazine derivative 133. Then, deprotection of the methyl ether 133 occurred to give favipiravir (134).

8. Conclusion

Most of the extraordinary potential of fluorine-containing molecules, as FDA-approved antiviral agents in medicinal chemistry and drug discovery, has been covered and clarified in this review. Clearly, the introduction of a fluorine atom or trifluoromethyl group to these drugs enhances both their pharmacodynamic and pharmacokinetic properties. Future directions must include larger clinical trials to assess the safety, tolerability, and resistance of the antiviral drugs. The antiviral combination strategy is highly beneficial to solve problems with these drugs.

Conflicts of interest

There are no conflicts of interest to declare.

Acknowledgements

M. S. A. express sincere gratitude to Academy of Scientific Research and Technology (Egypt) for funding project no 7207.

References

1. E. De Clercq, Biochim. Biophys. Acta, Mol. Basis Dis., 2002, 1587, 258–275.
2. D. R. Kuritzkes, Curr. Opin. HIV AIDS, 2009, 4, 82–87.
3. E. F. Tchesnokov, C. Gilbert, G. Boivin and M. Götte, J. Virol., 2006, 80, 1440–1450.
4. E. De Clercq, J. Clin. Virol., 2004, 30, 115–133.
5. D. J. Hazuda, P. Felock, M. Witmer, A. Wolfe, K. Stillmock, J. A. Grobler, A. Espeseth, L. Gabryelski, W. Schleif and C. Blau, Science, 2000, 287, 646–650.
6. P. A. Furman, J. A. Fyfe, M. H. St Clair, K. Weinhold, J. L. Rideout, G. A. Freeman, S. N. Lehrman, D. P. Bolognesi, S. Broder and H. Mitsuya, PNAS, 1986, 83, 8333–8337.
7. E. Tuaillon, M. Gueudin, V. Lemée, I. Gueit, P. Roques, G. E. Corrigan, J.-C. Plantier, F. Simon and J. Braun, J. Acquired Immune Defic. Syndr., 2004, 37, 1543–1549.
8. J. Martinez-Picado, M. P. DePasquale, N. Kartsonis, G. J. Hanna, J. Hong, D. Finzi, E. Rosenberg, H. F. Günthard, L. Simon and A. Savara, PNAS, 2000, 97, 10948–10953.
9. E. De Clercq, Nat. Rev. Drug Discovery, 2006, 5, 1015–1025.
10. S. Mahmoud, S. Hasabelnaby, S. Hammad and T. Sakr, J. Adv. Pharm. Res., 2018, 2, 73–88.
11. P. Shah and A. D. Westwell, J. Enzyme Inhib. Med. Chem., 2007, 22, 527–540.
12. T. Al-Harthy, W. Zoghaib and R. Abdel-Jalil, Molecules, 2020, 25, 4677.
13. K. L. Kirk, J. Fluorine Chem., 2006, 127, 1013–1029.
14. M. Derudas and S. Meneghesso, in Fluorinated Pharmaceuticals: Advances in Medicinal Chemistry, Future Science Ltd, 2015, pp. 30–42.
15. A. V. Bondi, J. Phys. Chem., 1964, 68, 441–451.
74 D. M. Volochnyuk, O. O. Grygorenko and A. O. Gorlova, Fluorine in Heterocyclic Chemistry, 2014, vol. 2, pp. 577–672.
75 C. Tantillo, J. Ding, A. Jacobo-Molina, R. G. Nanni, P. L. Boyer, S. H. Hughes, R. Pauwels, K. Andries, P. A. J. Janssen and E. Arnold, J. Mol. Biol., 1994, 243, 369–387.
76 J. Ding, K. Das, H. Moereels, L. Koymans, K. Andries, P. A. J. Janssen, S. H. Hughes and E. Arnold, Nat. Struct. Biol., 1995, 2, 407–415.
77 K. Das, J. Ding, Y. Hsiou, A. D. Clark Jr, H. Moereels, L. Koymans, K. Andries, R. Pauwels, P. A. J. Janssen and P. L. Boyer, J. Mol. Biol., 1996, 264, 1085–1100.
78 O. J. D’Cruz and F. M. Uckun, J. Antimicrob. Chemother., 2006, 57, 411–423.
79 S. R. Rabel, S. Sun, M. B. Maurin and M. Patel, AAPS PharmSci, 2001, 3, 26–29.
80 M. R. P. Rao and L. S. Babrekar, Indian J. Pharm. Sci., 2018, 80, 1115–1124.
81 S. Purser, P. R. Moore, S. Swallow and V. Gouverneur, Chem. Soc. Rev., 2008, 37, 320–330.
82 M. K. Gurjar, A. A. Deshmukh, S. S. Deshmukh and S. R. Mehta, US Pat., 8604189B2, 2013.
83 U. Batra, R. J. Higgins, K. C. Thompson and A. V. Kadtare, WO1999061026A1, 1999.
84 A. Cha and A. Budovich, Pharmacol. Ther., 2014, 39, 345.
85 V. K. Kaushik, V. K. Ravi, S. Vakiti and B. Tirumalaraju, WO2015097605A1, 2015.
86 https://www.hepatitisc.uw.edu, accessed 23rd August 2022.
87 K. M. Bullard-Feibelman, J. Govero, Z. Zhu, V. Salazar, M. Veselinovic, M. S. Diamond and B. J. Geiss, Antiviral Res., 2017, 137, 134–140.
88 R. W. Scott, J. P. Vitale, K. S. Matthews, M. G. Teresk, A. Formella and J. W. Evans, US Pat., 9056860B2, 2013.
89 R. W. Scott, F. Wang, B. Shi and E. Mogalian, WO2013184698A1, 2015.
90 D. V. Santi and T. T. Sakai, Biochemistry, 1971, 10, 3598–3607.
91 C. Heidelberger, D. G. Parsons and D. C. Remy, J. Med. Chem., 1964, 7, 1–5.
92 C. Heidelberger, D. Parsons and D. C. Remy, J. Am. Chem. Soc., 1962, 84, 3597–3598.
93 E. Vanderlinden, B. Vrancken, J. Van Houdt, V. K. Rajwanshi, S. Gillemot, G. Andrei, P. Lemey and L. Naesens, Antimicrob. Agents Chemother., 2016, 60, 6679–6691.
94 Y. Furuta, K. Takahashi, K. Shiraki, K. Sakamoto, D. F. Smee, D. L. Barnard, B. B. Gowen, J. G. Julander and J. D. Morrey, Antiviral Res., 2009, 82, 95–102.
95 E. De Clercq, Curr. Med. Chem., 2015, 22, 3866–3880.
96 T. Manabe, D. Kambayashi, H. Akatsu and K. Kudo, BMC Infect. Dis., 2021, 21, 1–13.
97 Q. Guo, M. Xu, S. Guo, F. Zhu, Y. Xie and J. Shen, Chem. Pap., 2019, 73, 1043–1051.
98 Y. Furuta and H. Egawa, WO2000010569A1, 2000.
99 E. J. Lane, Fujifilm in Avigan API license with Zhejiang Hisun Pharmaceuticals, Fierce Pharma, 22 June 2016, retrieved 22 September 2022.