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Multi-Channel Exploration of O Adatom on TiO$_2$(110) Surface by Scanning Probe Microscopy

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Abstract: We studied the O$_2$ dissociated state under the different O$_2$ exposed temperatures with atomic resolution by scanning probe microscopy (SPM) and imaged the O adatom by simultaneous atomic force microscopy (AFM)/scanning tunneling microscopy (STM). The effect of AFM operation mode on O adatom contrast was investigated, and the interaction of O adatom and the subsurface defect was observed by AFM/STM. Multi-channel exploration was performed to investigate the charge transfer between the adsorbed O and the TiO$_2$(110) by obtaining the frequency shift, tunneling current and local contact potential difference at an atomic scale. The tunneling current image showed the difference of the tunneling possibility on the single O adatom and paired O adatoms, and the local contact potential difference distribution of the O-TiO$_2$(110) surface institutionally revealed the charge transfer from TiO$_2$(110) surface to O adatom. The experimental results are expected to be helpful in investigating surface/interface properties by SPM.

Keywords: multi-channel; surface property; scanning probe microscopy

1. Introduction

Scanning probe microscopy (SPM) has developed as a powerful tool for exploring the surface properties and surface dynamic process at an atomic scale on a semiconductor or insulator [1–10]. For example, atomic manipulation has been realized, and surface chemical reactions have been observed with atomic resolution by atomic force microscopy (AFM). Based on AFM, Kelvin probe force microscopy (KPFM) was developed to characterize the contact potential difference (CPD) between the substrate and cantilever tip. CPD originates from the difference of the work functions and is specifically referred to as the local CPD (LCPD) in atomic-resolution KPFM [11,12]. To date, different modes of KPFM have been successfully used to simultaneously measure surface structures and LCPD, and the surface potential of TiO$_2$(110) was measured [13–19]. Local density of states (LDOS) gives significant information of the electronic structure of the surface, which is measured by scanning tunneling microscopy (STM). Combining AFM/STM techniques has been developed to investigate the surface structure and LDOS [20–22]. Simultaneous measurement of tunneling current and LCPD is useful to explore the surface properties and surface reaction process, but it cannot be achieved by the conventional KPFM due to the regulation of DC bias voltage. To simultaneously characterize the electronic structure and LCPD, we previously proposed a method to achieve the frequency shift ($\Delta f$), average tunneling current ($\langle I_t \rangle$) and LCPD by the KPFM without DC bias voltage feedback, and this method was successfully performed on the rutile TiO$_2$(110) surface [19]. Hence, based on
AFM, multi-channel exploration has the potential to explore the charge transfer of the interface and to unravel physical features for understanding the surface catalytic process.

Oxygen molecular interaction with TiO$_2$ has attracted much attention and has become a prototypical model due to its importance in many surface reactions. Molecular oxygen acts as the main oxidizing reagent in many catalytic reactions and is used as an electron scavenger, which is believed to facilitate surface reactions. The adsorption and dissociation of O$_2$ on the rutile TiO$_2$(110) surface at low and room temperatures have been well characterized by STM experimental and theoretical study, where the oxygen molecular adsorbed on the Ti$_5$C site or the surface bridging oxygen vacancies (O$_v$) and the oxygen adatom (O$_{ad}$) along the fivefold-coordinated Ti$_5$C site is formed [23,24]. Importantly, it is believed that chemical adsorption and dissociation of O$_2$ on the TiO$_2$ surface accompanied by the charge transfer and the activation of O$_2$ dissociation are key factors in the reaction process, which is importantly related to their charge state. Therefore, multi-channel exploring of the state of O$_{ad}$ on TiO$_2$(110) surface is useful in understanding its fundamental mechanism.

In this study, we first showed the dissociation state of O$_{ad}$ on the TiO$_2$(110) surface under the different O$_2$-exposed temperatures by AFM, and then we characterized the O$_{ad}$ by simultaneous AFM/STM, which would provide complementary information in recording the topographic and tunneling current signals. Finally, simultaneous measurements of frequency shift, tunneling current and LCPD images on the O-TiO$_2$(110) surface are taken by SPM. The tunneling current contrast and surface potential difference were analyzed to explore the charge transfer between the O$_{ad}$ and TiO$_2$ surface. The experimental results and methods are useful in characterizing the other nanomaterials.

2. Experimental Details

Experiments were performed with a home-built no contact (NC)-AFM system under ultrahigh vacuum conditions ($3 \times 10^{-11}$ Torr) at 78 K, which was operated in frequency modulation (FM). An AFM cantilever was oscillated at a constant amplitude and at its resonant frequency by automatic gain control (AGC). AFM/STM simultaneous measurements can be carried out in two ways, and force signal and tunneling current are recorded in the separated channel. When the frequency shift was operated as the feedback signal, topographic and $<I_t>$ images were simultaneously recorded and $<I_t>$ contained crosstalk of tip motion. When measurements were taken in the constant height mode, frequency shift and $<I_t>$ could be simultaneously obtained. In this mode, two signals were independently measured, so they did not contain the artificial signal. The comparison of two operation modes will be shown in the results section. Figure 1 shows the experimental setup for simultaneous measurements of topography, $<I_t>$ and LCPD. Topography and $<I_t>$ images were obtained by AFM, and LCPD images were recorded by FM-KPFM in the constant height mode. The equation for $V_{LCPD}$ was derived as follows; details can be found in Ref. [18,19]:

$$V_{LCPD} = V_{DC} - \text{sgn}(\alpha_m) \frac{V_{ac}}{4} \left| \Delta f_m \right| T(f_{2m}) \left( \frac{\Delta f_m}{\Delta f_{2m}} \right)$$

Here, $V_{DC}$ is the dc voltage. The parameter $\text{sgn}(\alpha_m)$ is known by the phase difference between $V_{ac}$ and $|\Delta f_m|$. $|\Delta f_m|$ and $|\Delta f_{2m}|$ are the $f_m$ and $f_{2m}$ components of $\Delta f$, respectively. The $T(f_m)$ and $T(f_{2m})$ are the transfer functions of the phase-locked-loop (PLL).

The signal frequency shift was obtained by the phase locked loop (PLL) and divided into two parts. One was applied to adjust the tip–sample interaction with a band elimination filter (BEF), and the other was connected to FM-KPFM by feeding it into the lock-in amplifiers. An ac bias voltage was obtained by an oscillator and acted as the reference signal. The $f_m$ and $f_{2m}$ components of the frequency shift were detected by two lock-in amplifiers. As shown in Figure 1, $<I_t>$ was recorded in a separate channel from the tip to cancel the crosstalk, and bias voltage was applied to the sample.

The commercial Ir-coated cantilever (Nanosensors SD-T10L100, $f_0 \sim 800$ kHz) was used in the current study. The cantilever tip was first degassed at approximately 650 K for 30 min and then cleaned.
by Ar ion bombardment to remove the contaminants, prior to the measurements. Features of the surface structure were related to the charge states of the tip apex, and a stable tip was essential to accurately characterize the surface structure and properties in the experiment \[25,26\]. The imaging mode became stable in AFM experiments when the metal-coated Si cantilever was employed in the experiments \[27–29\].

Figure 1. Experimental setup of AFM/FM-KPFM for simultaneous measurements of topography, \(<I_t>\) and LCPD.

The TiO\(_2\)(110) sample surface (provided by Furuuchi Chemical Corporation, Hyogo, Japan) was prepared by several cycles of Ar ion sputtering and subsequent annealing at 1000 K for 20 min. After that, the freshly cleaned surface was exposed to O\(_2\) in the preparation chamber and then transferred into the observation chamber. AFM images were taken after the sample temperature decreased to 78 K.

3. Results and Discussion

We first introduced the surface structure model of rutile O-TiO\(_2\) (110)-(1 × 1) and O\(_2\) dissociation state at room temperature (RT) and 400 K. Figure 2a shows a ball model of the rutile O-TiO\(_2\) (110)-(1 × 1) surface, which consists of alternating Ti\(_{5c}\) rows and sixfold-coordinated Ti\(_{6c}\) rows surrounded by in-plane threefold-coordinated O\(_{3c}\) rows and bridging twofold-coordinated O\(_{2c}\) rows. The single O\(_{ad}\) (O\(_{ad}\): light green ball) formed by O\(_2\) dissociation at the O\(_v\) site indicated that one O atom healed O\(_v\) and the other located at the Ti\(_{5c}\) site, and paired O\(_{ad}\) resulted from O\(_2\) dissociation at the Ti\(_{5c}\) site.

As reported in the previous literature \[30–32\], atomic contrast in the AFM image depended on the tip apex polarity, and surface defects were used as markers to distinguish the imaging mode. Hole and protrusion modes usually appeared in the imaging modes. When the tip apex was positively charged, the O\(_{2c}\) row was bright on the image due to the larger attractive force between the tip and the negative O\(_{2c}\) row. Surface defects were imaged as dark holes, which is called the hole mode \[28,29\]. When the tip apex was negatively charged, the contrast was inverted compared with that in the hole mode, and H atoms appeared as brighter spots than the O\(_v\) defects, which is called the protrusion mode. Figure 2b shows the topography image of the O-TiO\(_2\)(110)-(1 × 1) surface recorded in the hole mode. According to the experiment, bridging O\(_{2c}\) and Ti\(_{5c}\) rows were imaged as bright and dark features, respectively, and the bright spots on the Ti\(_{5c}\) rows are O\(_{ad}\). As introduced before, the single O\(_{ad}\) (denoted by the white dotted circle) was attributed to O\(_2\) dissociation at the O\(_v\) site, and paired O\(_{ad}\) separated by three lattice distances (denoted by the white dotted oval circle) was the result of O\(_2\) dissociation at the Ti\(_{5c}\) site.
was at or beyond RT, and results are consistent with the conventional STM observations [24,33]. Next, O behavior when the O charge on these paired O significantly, depending on the level of bulk reduction. Current experiments suggest that the excess mechanism [34]. The concentration and distribution of Ti at RT, and the corresponding line profiles along the O ad. The contrast is the same as Figure 2b. O2c and Ti5c rows are imaged as bright and dark features, respectively, and the bright spots on the Ti5c rows are O ad. Here, paired O ad separated by one lattice constant is denoted as the P-O ad(1). As shown in Figure 3a,b and Figure 2b, single O ad, P-O ad(1), P-O ad(2) and P-O ad(3) are observed when O2 is exposed to the TiO2 surface at room temperature (RT). As introduced before, single O ad was attributed to O2 dissociation at the Ov site, and paired O ad was the result of O2 dissociation at the Ti5c site. In our results, a single O ad atom was the distinctly dominant state of O ad, when the exposure temperature was at RT. P-O ad(2) is the second preferred state. Density functional theory (DFT) showed the P-O ad(2) configuration was the most preferred structure at RT, and further O ad diffusion (P-O ad(2) to P-O ad(3)) was hindered by a barrier of 1.3 eV [24]. In addition, the separation of P-O ad(1) to P-O ad(2) was exothermic by 0.4 eV theoretically. P-O ad(1) and P-O ad(3) configurations were observed in our experiments, but they were rare because separation of O ad is the result of a balance between Coulombic repulsion of two O ad and is thermally driven, and P-O ad(1) and P-O ad(3) configurations can be generated.

When the sample was exposed to O2 conditions beyond 350 K, dissociated O2 could overcome the diffusion barrier forming the P-O ad(3) structure [33]. The number of paired O ad distinctly increased on the O2-exposed surface at 400 K, and the states of the paired O ad were mainly P-O ad(3) and P-O ad(5) configurations, shown in Figure 3c,d. Under high temperatures for the O2-exposed surface (>400 K), the Ti interstitials (Ti int) can diffuse from the bulk to the near-surface region via an interstitial diffusion mechanism [34]. The concentration and distribution of Ti int in the near-surface region can vary significantly, depending on the level of bulk reduction. Current experiments suggest that the excess charge on these paired O ad is mainly provided by the Ti int rather than Ov in determining the adsorption behavior when the O2-exposed surface temperatures went beyond 400 K. Our results demonstrate that O2 dissociatively adsorbed on the rutile TiO2(110)-(1 × 1) surface when the temperature of O2 exposure was at or beyond RT, and results are consistent with the conventional STM observations [24,33]. Next, we explored the O ad on TiO2 surface by AFM/STM.

Figure 2. (a) Ball model of the rutile O-TiO2(110)-(1 × 1) surface. (Ti5c: orange balls, O5c: dark blue balls, O2c: gray balls, O ad: light green balls) (b) AFM image of O-TiO2(110)-(1 × 1) surface. Bright and dark rows are O2c and Ti5c rows, respectively, and bright spot is O ad. $f_0 = 807$ kHz, $Q = 23620$, $\Delta f = -70$ Hz, $V_{DC} = 0.6$ V and $A = 500$ pm, image size: $6.3 \times 5.3$ nm $^2$.

Figure 3 shows two AFM topographic images of the rutile O-TiO2(110)-(1 × 1) surface exposed to O2 at RT and 400 K, respectively, and the corresponding line profiles along the O ad. The contrast in the AFM image depended on O2 and Ti int. As reported in the previous literature [30–32], atomic contrast in the AFM image depended on the tip apex polarity, and surface defects were used as markers to distinguish the imaging mode.
AFM adsorbate and substrate, and constant height operation is necessary. Next, we investigated the O not repulsive to O observed in Figure 4d, and they are O in constant height mode. A bright spot and two weak bright spots (denoted by the white square) are in the image. The bright spot marked by an oval circle is due to subsurface defects, which was not probed in the crosstalk of tip motion. When the tip moves on the O ad site, the additional attractive force acts in the tip–sample interaction, and the tip has to retract in order to keep a constant frequency shift. Thus, tunneling current dramatically decreases, and the contrast of Oad becomes a dark spot in the topographi c image. Usually, the empty state is imaged at positive sample bias voltage in STM, so O2c and Ti5c rows are imaged as dark and bright rows, respectively, where the contrast of Oad in Figure 4d is reversed compared with the topographic image. The dark spot marked by the white dashed circle is Oad, and this is different from the conventional STM image of O-TiO2(110), which is caused by the crosstalk of tip motion. When the tip moves on the Oad site, the additional attractive force acts in the tip–sample interaction, and the tip has to retract in order to keep a constant frequency shift. Thus, tunneling current dramatically decreases, and the contrast of Oad becomes a dark spot in <I< image. The bright spot marked by an oval circle is due to subsurface defects, which was not probed in Figure 4a, ever reported by our group or other groups [19,22]. Figure 4c is the ∆f image and Figure 4d is the corresponding tunneling current image, recorded in the constant height mode. Compared with Figure 4a,b, the image contrast is the same, except the Oad in Figure 4d. Oad is imaged as bright spot in the <I< image due to elimination of the crosstalk between topography and <I< signals in the constant height mode. A bright spot and two weak bright spots (denoted by the white square) are observed in Figure 4d, and they are Oad and subsurface defects. It indicates the subsurface defect is not repulsive to Oad. Therefore, AFM/STM is a useful technique to investigate the interaction of the adsorbate and substrate, and constant height operation is necessary. Next, we investigated the Oad by AFM/STM/KPFM.

Figure 3. (a) AFM images on TiO2(110) surface. (f0 = 802 kHz, Q = 13001, ∆f = −80 Hz, VDC = 0.4 V, A = 500 pm, 5 x 5 nm2, O2 exposure at RT). (b) Line profiles along the oxygen adatoms in (a). (c) AFM images on TiO2(110) surface. (f0 = 795 kHz, Q = 6329, ∆f = −231 Hz, VDC = 0.4 V, A = 500 pm, 4.5 x 5 nm2, O2 exposure at 400K). (d) Line profiles along the oxygen adatoms in (c). The four images in Figure 4 are obtained in the same area of the O-TiO2(110) surface. Figure 4a and 4b experimentally show simultaneously recorded topographic (Z) and <I< images recorded in the constant frequency shift mode. In the topographic image (Figure 4a), the atomic contrast is the same as that in Figure 3a, in that the bright and dark rows are the O2c and Ti5c rows, respectively. A bright spot marked by the dashed white circle is Oad. Figure 4b demonstrates the corresponding tunneling current image. Usually, the empty state is imaged at positive sample bias voltage in STM, so O2c and Ti5c rows are imaged as dark and bright rows, respectively, where the contrast of O2c and Ti5c rows is reversed compared with the topographic image. The dark spot marked by the white dashed circle is Oad, and this is different from the conventional STM image of O-TiO2(110), which is caused by the crosstalk of tip motion. When the tip moves on the Oad site, the additional attractive force acts in the tip–sample interaction, and the tip has to retract in order to keep a constant frequency shift. Thus, tunneling current dramatically decreases, and the contrast of Oad becomes a dark spot in <I< image. The bright spot marked by an oval circle is due to subsurface defects, which was not probed in Figure 4a, ever reported by our group or other groups [19,22]. Figure 4c is the ∆f image and Figure 4d is the corresponding tunneling current image, recorded in the constant height mode. Compared with Figure 4a,b, the image contrast is the same, except the Oad in Figure 4d. Oad is imaged as bright spot in the <I< image due to elimination of the crosstalk between topography and <I< signals in the constant height mode. A bright spot and two weak bright spots (denoted by the white square) are observed in Figure 4d, and they are Oad and subsurface defects. It indicates the subsurface defect is not repulsive to Oad. Therefore, AFM/STM is a useful technique to investigate the interaction of the adsorbate and substrate, and constant height operation is necessary. Next, we investigated the Oad by AFM/STM/KPFM.
Vad was 52 mV, as shown in Figure 6b. This indicates the charge state of O ad profile (not shown here), and this was in good agreement with our previous studies [18,19].

very pronounced due to the combined e distribution was also di
ff between the Oad and Ti5c row was about 75 mV. It intuitively suggests the electrons transferred from TiO2 to Oad.

Figure 5 shows frequency shift, tunneling current and local contact potential difference images with atomic resolution and corresponding line profiles along the single Oad. In the experiment, the measurement was performed in the constant height mode to prevent crosstalk between the signals of the frequency shift and tunneling current. In the Δf image shown in Figure 5a, O2c and Ti5c rows are simultaneously observed as bright rows with super high resolution, and Oad is imaged as the bright spot. In the <I> image (see Figure 5b), the contrast was the same as in Figure 5a, except O rows are imaged as dark, which is consistent with the previous studies by conventional STM in that the conduction band of TiO2 is dominated by Ti 3d states, and bright features are usually assigned to the empty Ti 3d states of Ti5c ions under positive sample bias voltage, even though they lie lower than the bridging O2c rows [35]. The tunneling current value on Oad was higher than that on the Ti5c rows, and the current difference between the Oad and Ti5c row was about 0.375 nA. The contrast in STM images depends on the different contributions of Ti 3d and O 2p states and their different decay as a function of the tip–sample separation [36]. Oad is higher than O2c and Ti5c rows in surface geometry, and Oad finally appeared as bright on the image. It was clearly observed that there was some depression around the Oad, and this was due to the decreased numbers of empty states near the conduction band for tunneling, which resulted from the negatively charged Oad [24]. This phenomenon was more pronounced in paired Oad, as shown in the following tunneling current line profile. In the V LCPD image (see Figure 5c), the imaging contrast was the reverse of that in the image, except for the Oad. The relative value of V LCPD between the Ti5c and O2c sites was approximately 28 mV from the line profile (not shown here), and this was in good agreement with our previous studies [18,19]. V LCPD had a higher value on the Oad than on the proximate Ti5c rows, and the relative value of CPD between the Oad and Ti5c row was about 75 mV. It intuitively suggests the electrons transferred from TiO2 to Oad.

As shown in Figure 6, line profiles are plotted along the paired Oad. The current difference between the Oad and adjacent Ti5c row was about 0.5 nA. The depression between the paired Oad was very pronounced due to the combined effect in the corresponding line profile. The surface potential distribution was also different around the single O adatom and paired Oad. The potential difference was 52 mV, as shown in Figure 6b. This indicates the charge state of Oad is different, and excess electrons around the paired Oad are shared by the paired Oad resulting in that the relative value of

Figure 4. (a,b) Topographic and corresponding tunneling current images of O-TiO2(110) surface obtained in the constant frequency shift mode. There is crosstalk between topography and <I> signals and Oad is imaged dark spot in <I> image. (c,d) frequency shift and corresponding tunneling current images of O-TiO2(110) surface obtained in the constant height mode. The crosstalk between topography and <I> signals is eliminated.
CPD between the O\textsubscript{ad} and Ti\textsubscript{5c} row is lower. Therefore, multi-channel exploration demonstrates the powerful ability to explore surface properties with SPM.

**Figure 5.** Multiple images of TiO\textsubscript{2}(110) surface with atomic resolution and corresponding line profiles. (a) Frequency shift (\Delta f) image, (b) tunneling current (<I\textsubscript{t}> image and (c) local contact potential difference (V\textsubscript{LCPD}) image. (d,e) The line profiles along the blue line on the surface in (b,c). (f_0 = 805 kHz, Q = 27623, \Delta f = -260 Hz, V\textsubscript{DC} = 1.3 V, V\textsubscript{AC} = 1.5 V, A = 500 pm, size: 3.5 \times 3.2 nm\textsuperscript{2}).

**Figure 6.** The line profiles along the dark line on the surface in Figure 5b,c. (a) Tunneling current and (b) local contact potential difference line profiles along the paired O adatoms.

### 4. Conclusions

We studied the O\textsubscript{2} dissociated state under different O\textsubscript{2}-exposed sample surfaces with high resolution, and we investigated the electron charge transfer between the adsorbed O and TiO\textsubscript{2} substrate with multi-channel exploration. We observed the interaction between O\textsubscript{ad} and the subsurface, and we confirmed the electron charge transfer from the TiO\textsubscript{2} surface to the adsorbed O upon O\textsubscript{2} dissociation on the TiO\textsubscript{2}(110) surface by tunneling current and local contact potential difference. Our results demonstrated that multi-channel exploration was able to obtain the surface structures and charge transfers between the adsorbate and substrate, and this is expected to be useful for investigating the surface properties and charge transfer phenomena at the interface.

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