Antioxidant Capacity and Polyphenolic Compounds of Blackberries Produced in Different Climates

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Abstract. Antioxidants are compounds with varied chemical structures that are affected by biotic and abiotic factors. The objective of this study was to characterize and compare bioactive compounds and the antioxidant capacity of fruit from four blackberry cultivars produced under different climatic conditions. Ascorbic acid content, total polyphenols, flavonoids, monomeric anthocyanins and antioxidant activity of the fruit were evaluated, and high levels of bioactive compounds as well as antioxidant activity were observed regardless of the cultivar or growing location. The results showed that bioactive production is affected by the cultivar and environment. Furthermore, the antioxidant potential of the blackberry fruit depends on the total phenolics and anthocyanin.

Several studies have indicated that the ingestion of natural antioxidants found in fruits and vegetables reduces the damage caused by free radicals and protects the body against many diseases (Schulz et al., 2019; Souza et al., 2014; Van de Velde et al., 2016). However, industrial research is facing the important challenge of substituting synthetic antioxidant compounds for substances derived from plants. Therefore, there is an emphasis on products that fulfill such demands, and it is necessary to find potential sources of such compounds in nature and elucidate the processes related to their production in plants (Crote et al., 2019). Blackberry presents itself as one of the most promising species due to its high potential of the blackberry fruit depends on the total phenolics and anthocyanin. in blackberries compared with 12 other species of small fruits.

The amount and distribution of antioxidant compounds vary substantially among different blackberry cultivars, as shown by divergent published data regarding the content and classes of compounds found in blackberries (Mullen et al., 2002; Paredes-López et al., 2010). It is inferred that biotic and abiotic factors such as genetics, climate, water availability, and crop management have an important role in the level of bioactive compounds and antioxidant capacity of blackberries (Castrejón et al., 2008; Reyes-Carmona et al., 2005).

According to Strik (2008), there are more than 20,000 hectares of blackberry worldwide, and there are many cultivars (Campagnolo and Pio, 2012; Clark and Finn, 2011). Some of these were introduced to Brazil in the 1970s, when a breeding program was launched to attend the demands of some growers. Among them are Tupy, the most widely grown cultivar in South America, and Xavante, a thornless, very promising cultivar in terms of productivity (Crote et al., 2016; Pio and Gonçalves, 2014).

Therefore, the objective of this research was to characterize and compare bioactive compounds and the antioxidant activity of four blackberry cultivars grown under different climates (temperate and humid mesothermal climate).

Materials and Methods

Tupy, Guaraní, Xavante, and Cherokee blackberry cultivars were planted in three municipalities of Parana State in Brazil (Lapa, Pinhais, and Cerro Azul) in 2011. The first two, according to the Koppen climate classification, were planted under a temperate climate (Cfb) and the third was planted under a humid mesothermal climate (Cfa). The Koppen climate classification defines a climate as “temperate” when the mean temperature is higher than – 3 °C (26.6 °F) but lower than 18 °C (64.4 °F) during the coldest month. In contrast, the humid subtropical climates have a warm and wet flow from the tropics that creates warm and moist conditions during the summer months. As such, summer is often the wettest season (Fig. 1).

In Lapa, blackberries were drip-irrigated with 25 mm of water per week. Aside from this, the cultural practices were the same for the three areas, with the pH adjusted to 5.5 and fertilizer based on the recommendation of Nagaguma and Clark (1997). A completely randomized experimental design with four replications of six plants per cultivar (both border plants were disregarded) was applied. Plants were conducted on V-shape trellis with three wires on both sides. Fruits were harvested during their physiological maturation stage (Souza et al., 2014). In the laboratory, a completely randomized design with four subdivided parcels was used. Four replications, four cultivars, three locations, and 25 fruits per treatment were used.

The following chemicals, which were obtained from Sigma Aldrich (São Paulo, Brazil), were used in the experiments: 2,2-azino-bis(3-ethylbenzthiazoline-6-sulphonic acid) (ABTS); aluminum chloride; catechin; hydrochloric acid; 2,2-diphenyl-1-picrylhydrazyl (DPPH); ethanol; Folin–Ciocalteu reagent; gallic acid; methanol; potassium sulfate; potassium persulfate; sodium carbonate; sodium hydroxide; sulfuric acid; and oxalic acid.

Fruits were stored under refrigeration until extraction and held on the same day of harvest. Juice was extracted with a domestic centrifugal juicer and treated with acidified methanol (HCl 0.01%) for phenolic compound (total polyphenols, flavonoids, and anthocyanins) and antioxidant activity analyses.

To determine the ascorbic acid content of the samples, the AOAC International (1992) method was used, but with the standard extraction solution (methaphosphoric acid solution) substituted for oxalic acid solution. To quantify vitamin C, a standard ascorbic acid curve was used. Results were expressed in mg of ascorbic acid per 100 g of fresh fruit (FF).

The anthocyanin content was determined by the pH difference method (Giust and Wroslad, 2001), whereby the extracts are dissolved in two buffer systems: potassium chloride pH 1.0 (0.025 M) and sodium acetate pH 4.5 (0.4 M). Absorbance of the extracts was measured at 510 and 700 nm.
Results were expressed in mg of monomeric anthocyanin per 100 g of FF. The total polyphenol content was determined by the Folin and Ciocalteu method, whereby the mix of phosphor wolframic and phospholipid acids in the basic medium reduces itself while oxidizing the phenolic compounds, creating blue wolframic oxides (\(W_8\)O\(_{23}\)) and molybdenum (\(Mo_8\)O\(_{23}\)). Readings were performed in a spectrophotometer at a wavelength of 725 nm (Moyer et al., 2002). The results were expressed in mg of gallic acid equivalent (GAE) in 100 g of FF using a prepared standard curve with solutions of gallic acid in concentrations ranging from 0 to 600 mg L\(^{-1}\).

The total flavonoid content in the fruit was quantified by the spectrophotometer method at a wavelength of 425 nm (Zhishen et al., 1999). The results were expressed in mg of quercetin per 100 g of FF by a standard curve prepared with quercetin solutions in concentrations ranging from 0 to 300 mg L\(^{-1}\).

One of the methods used to quantify antioxidant activity was the DPPH free radical scavenging method, which was in accordance with the methodology of Brand-Williams et al. (1995) and modifications by Maro et al. (2013). The kinetic reaction between the antioxidants present in the sample and the DPPH radical was registered in three different concentrations to graphically obtain the EC\(_{50}\), i.e., the fresh weight concentration needed to capture 50% of the radicals initially present in the sample. The absorbance was recorded at 515 nm.

The degradation analysis of the ABTS free radical by the antioxidants present in the sample followed the method proposed by Re et al. (1999). Absorbance was read in 734 nm for 7 min and used to estimate the scavenging capacity of the sample free radicals by the following equation: % of ABTS radical scavenging = \(\left[1 - \left(\frac{\text{Abs } t}{\text{Abs control}}\right)\right] \times 100\), where Abs \(t\) is the sample absorbance at a given time of the analysis and Abs control is the sample absorbance at time zero.

An analysis of variance (ANOVA) was performed; when variables were significant (\(P \leq 0.05\)), the Tukey test was used for multiple comparisons. To visualize the relationship of the chemical properties with the antioxidant capacity of the blackberry fruits, correlation graphs were plotted. SAS (SAS Institute 1999; Cary, NC) and StatSoft, Inc. (2007) software packages were used for the statistical analyses.

**Results and Discussion**

Ascorbic acid, total polyphenol, flavonoid, and anthocyanin contents are presented in Table 1. A variance analysis showed significant interactions between the cultivar and environment in the production of the blackberry active compounds.

Results showed significant levels of ascorbic acid in *in natura* blackberry, with an average of 24.5 mg per 100 g of fruits, i.e., 17% of the daily recommended intake for humans. However, this fruit cannot be recommended as the only dietary source of this.

![Fig. 1. Monthly minimum, average, and maximum temperatures (°C) and cumulative rainfall (mm) in Pinhais, Lapa, and Cerro Azul in PR. Minimum temperature (black bar); average temperature (white bar); maximum temperature (shaded bar); and cumulative rainfall (x). Simepar Weather Station data.](image-url)
Table 2. Antioxidant activity of four blackberry cultivars grown under different climatic conditions (Lapa, Pinhais, and Cerro Azul) found by the DPPH (EC50) and ABTS (% of radical degradation) methods.

| Cultivar | Lapa, PR | Pinhais, PR | Cerro Azul, PR |
|----------|----------|-------------|----------------|
| Tupy     | 1.74 bC  | 3.08 aA     | 2.45 aB        |
| Guarani  | 1.74 bb  | 2.18 bb     | 2.17 bb        |
| Xavante  | 0.86 cc  | 2.12 bb     | 1.58 cc        |
| Cherokee | 1.85 aC  | 2.27 bb     | 2.19 bb        |

Means in the same column followed by the same small or capital letter do not differ significantly (P ≤ 0.05) according to Tukey’s test.

Ascorbic acid is a hydrosoluble vitamin that is easily oxidized by heat, light, and oxygen. Under humid subtropical conditions, there are higher temperatures, which may have contributed to the degradation of this vitamin in the fruits. There were variations among cultivars and growing locations; however, no observed pattern was able to indicate which cultivars and locations were better for ascorbic acid production. The levels varied from 20.4 to 28.1 mg/100 g of FF. Pantelidis et al. (2007) found great variations in the levels of vitamin C in different Rubus cultivars (14.3 and 103.3 mg/100 g of FF). Deighton et al. (2000) found values between 12.3 and 16.4 mg/100 g FF in wild blackberry species.

In Table 1, variations in total polyphenol concentrations according to climate and cultivar are shown. Moyer et al. (2002) found values of 275 to 678 mg of GAE/100 g FF in a group of 27 blackberry hybrids. The values found herein were higher than those found in strawberries (Isabelle et al., 2010; Kusakoski et al., 2006). Souza et al. (2014) also found higher levels of total polyphenols in blackberry compared with other small fruits, such as strawberry and raspberry, grown under tropical conditions. It was observed that fruits harvested in Lapa, PR (temperate climate, under irrigation), presented higher levels of total polyphenols (Table 1). Conversely, the humid mesothermal climate of Cerro Azul, PR, induced lower production of these compounds. Chemically, total polyphenols are synthesized mainly through the shikimate pathway in plants, and the main enzyme is the ammonium phenylalanine lyase (PAL). The action of this enzyme is regulated by environmental factors such as temperature and water availability (Cechinel-Filho, 2012; Manach et al., 2004), which explains the differences found in the fruits from the three locations.

Hydric stress promotes significant consequences for the concentration of secondary metabolites in plants. Many reports have inferred that such conditions increase the compound composition. However, regarding phenolic metabolites, these studies present conflicting results (Dustin and Cooper-Driver, 1992; Tattini et al., 2004). It is possible that the effect of drought on the concentrations of these metabolites is dependent on stress levels and the physiological development period of occurrence, with short periods related to greater productivity and long periods related to less productivity. Furthermore, the irregularity of rain distribution throughout the year may have caused a reduction in the polyphenol production of blackberry grown in Pinhais and Cerro Azul, which were not irrigated (Fig. 1).

Regarding temperature, a positive correlation between the intensity and duration of cold and mRNA producers of key enzymes of the shikimic acid pathway, such as PAL, has been reported (Cechinel-Filho, 2012). Therefore, the temperate climate in Lapa and Pinhais, because these areas present more rigorous winters, had a positive effect on polyphenol production (Fig. 1).

The Xavante cultivar presented higher total polyphenol production in its fruits (Table 1). This cultivar stands out because of its thorny stems, which is a recessive trait. However, its greater acidity of fruits is also a characteristic of this cultivar (Croge et al., 2016; Croge et al., 2019). This high acidity occurs due to a greater amount of tannins, which are also classified as

Table 3. Average Euclidean distance estimates of blackberry cultivar dissimilarities.

|          | Tupy | Guarani | Xavante | Cherokee |
|----------|------|---------|---------|----------|
| L        | P    | C       | L       | P        |
| Tupy     |      |         |         |          |
| P        |      |         |         |          |
| C        |      |         |         |          |
| Guarani  | 122  | 133     | 19      | 47       |
| Xavante  |      |         |         |          |
| Cherokee |      |         |         |          |

L = Lapa, PR; P = Pinhais, PR; C = Cerro Azul, PR.

vitamin. Souza et al. (2014) found higher levels of ascorbic acid in strawberries and blueberries (90.1 and 73.2 mg per 100 g of FF, respectively).

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polphenols. Hence, the higher contents found in this cultivar are justifiable (Croge et al., 2016; Fennema et al., 2010; Poling, 1996).

Fruits harvested in Lapa, PR, and from Xavante also presented higher concentrations of flavonoids and anthocyanins (Table 1). This is probably because flavonoids and anthocyanins are also shikimate pathway derivatives and dependent on PAL because they are a specific class of phenolic compound.

Regarding flavonoids, the average of 38.8 mg of quercetin obtained in 100 g of blackberry is considered high. Manach et al. (2004) reported that flavonoid levels are usually low in food (from 1.5 to 3.0 mg of quercetin per 100 g of FF), and that one of the richest sources of such compounds was blueberry, with concentrations ranging from 3 to 16 mg of quercetin per 100 g of FF. The concentrations in the blackberries analyzed herein exceeded those concentrations, suggesting that they are rich sources of such compounds. Souza et al. (2014) found flavonoid levels 1.8-times higher in blackberries than in blueberries. Studies have indicated that a diet rich in flavonoids may protect the body against cardiovascular diseases, neurodegenerative disorders, and some types of cancer (Fennema et al., 2010; Paredes-López et al., 2010).

The anthocyanin values found in blackberries were also high and exceeded the values described for other small fruits in the literature, thus emphasizing the great potential of blackberry fruits as natural color additives in the food, drug, and cosmetic industries (Table 1). Souza et al. (2014) found levels of 9.6, 38.2, and 47.5 mg/100 g of FF in raspberry, strawberry, and blueberry, respectively. Kuskoski et al. (2006) found 23.7 mg/100 g of FF in strawberry.

One of the most described properties of anthocyanin is its antioxidant activity (Martínez-Flórez et al., 2002; Nijveldt et al., 2001). Such activity may be confirmed by the high degradation level of ABTS radicals and low EC50 levels observed in blackberry fruits (Table 2).

The EC50 values were various and were lower in Xavante (i.e., fewer fruits are needed to degrade 50% of the free radicals) (Table 2). Furthermore, fruits from Lapa, PR, presented even higher antioxidant potential (lower EC50). However, even the lowest values indicated the higher antioxidant potential of blackberries compared with strawberries, raspberries, and blueberries (Souza et al., 2014).

Based on the degradation level of the ABTS free radical, Xavante presented the highest antioxidant potential (Table 2), and Tupy presented the lowest. Antioxidant activity measured by the ABTS method was similar in Tupy and Guarani cultivars from Lapa and Pinhais and different in Xavante and Cherokee cultivars. The temperate climate was more adequate for the production of antioxidant compounds than the humid mesothermal climate.

Souza et al. (2014) found that blackberries analyzed by ABTS present higher antioxidant concentrations than other small fruits, such as blueberries and strawberries. Therefore, this research has shown that blackberries are a good source of antioxidants and have potential use in various industries, especially Xavante fruits.

Fig. 3. Correlations between the bioactive compounds and antioxidant activity of blackberry fruits. (A) Total polyphenol content and antioxidant activity found by the ABTS method. (B) Total polyphenol content and antioxidant activity found by the DPPH method. (C) Monomeric anthocyanins and antioxidant activity found by DPPH. (D) Ascorbic acid and antioxidant activity found by ABTS.
Similarities among samples were evaluated using a hierarchical cluster analysis and Euclidean distances (Fig. 2, Table 3). The greatest divergence in the bioactive composition of fruits appeared between ‘Xavante’ from Lapa and ‘Tupy’ from Cerro Azul, followed by ‘Tupy’ from Pinhais. This further highlights the superiority of ‘Xavante’ and the influence of the agronomical conditions of Lapa in the bioactive composition of the fruits.

The antioxidant activity of blackberries had a positive correlation with the mono-meric anthocyanin and polyphenol contents (Fig. 3). A higher correlation was detected between the polyphenol content and antioxidant activity by the DPPH method \((R^2 = 0.935)\). The total polyphenol content was negatively correlated to EC50 values; hence, it was positively correlated to the antioxidant potential.

A high correlation was also perceived between the total polyphenol content and ABTS degradation \((R^2 = 0.816)\) and the monomeric anthocyanin content EC50 \((R^2 = 0.906)\). A positive correlation was also observed between monomeric anthocyanin and ABTS radical degradation. These data indicated that the total polyphenol content is responsible for the antioxidant potential of blackberry fruits, and that the higher the concentration of such compounds in the fruits, the better this potential will be.

Conclusion

Blackberry fruits are rich sources of phenolic compounds (total polyphenols, flavonoids, and anthocyanins) and have the potential to be used in industry as natural sources of such compounds. Levels of bioactive compounds and antioxidant activity of blackberry fruits are dependent on the cultivar and the climatic conditions of the growing location. Among the studied cultivars, Xavante has the greatest potential for industrial use. Furthermore, the plants are thornless, which facilitates cultural practices. A temperate climate allied with irrigation provided a better environment for the production of bioactive compounds in the blackberry cultivars studied. There was a high correlation between the bioactive compounds concentration and the antioxidant activity of the studied blackberry cultivars.

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