Supporting Information

for

Host–guest complexes of conformationally flexible C-hexyl-2-bromoresorcinarene and aromatic N-oxides: solid-state, solution and computational studies

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Experimental details, ¹H NMR solution-data, X-ray crystallography experimental details and computational data

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I General information

All solvents used for syntheses and crystal growth are reagent grade, and are used as received. Pyridine N-oxide (1), 2-methylpyridine N-oxide (2), 3-methylpyridine N-oxide (3), 4-methylpyridine N-oxide (4), 2,6-dimethylpyridine N-oxide (5), 4-methoxypyridine N-oxide (8), 4-phenylpyridine N-oxide (10), 4,4’-bipyridine N,N’-dioxide (11) and 4,4’-bipyridine N,N’-dioxide (12) as guests were purchased from Sigma Aldrich. C-ethyl-2-bromoresorcinarene (BrC2), C-propyl-2-bromoresorcinarene (BrC3) and C-hexyl-2-bromoresorcinarene (BrC6) were synthesized according to reported procedure [1]. 2-Methoxypyridine N-oxide (6), 3-methoxypyridine N-oxide (7) and 2,6-dimethoxypyridine N-oxide (9) were synthesized as reported [2]. The 1H NMR spectra were recorded on a Bruker Avance DRX 500 MHz spectrometer. Chemical shifts are calibrated to the residual solvent signals.

Ia General crystallization procedure: To a solution of BrC6 (0.018 mmol) in methanol (1.0 ml), was added respective aromatic N-oxide (0.070 mmol) dissolved in methanol (0.5 ml) at room temperature. The reaction mixtures were heated (3-5 minutes) to clear solutions using heat-gun and the hot solutions were filtered to remove insoluble precipitates. Slow evaporation of light orange-red colour filtrates provides single crystals suitable for X-ray diffraction analysis.

II Solid-state analyses

Single-crystal X-ray data for 3@BrC6, 4@BrC6, 6@BrC6, 7@BrC6, 8@BrC6, 11@BrC6, and 12@BrC6 were collected on a dual source Rigaku SuperNova Oxford diffractometer [3] equipped with an Atlas detector using mirror-monochromated Cu Kα (λ = 1.54184 Å) radiation. The data for 5@BrC6 were measured using a Rigaku SuperNova single-source Oxford diffractometer with an Atlas EoS CCD detector using mirror-monochromated Mo-Kα (λ = 0.71073 Å) radiation. Single-crystal X-ray data for BrC6 were measured on a Bruker Nonius Kappa CCD diffractometer [4] equipped with an APEX-II CCD detector using graphite-monochromated Mo Kα (λ = 0.71073 Å) radiation. The data obtained from Bruker Nonius Kappa diffractometer were performed using the program COLLECT [5] and HKL DENZO AND SCALEPACK [6]. The data collection and reduction for complexes performed using Rigaku instruments were done by the program CrysAlisPro [3], the gaussian face index absorption correction method [3] was used for these complexes. The intensities for data collected using Bruker Nonius Kappa diffractometer were corrected for absorption using SADABS [6] with multi-scan absorption correction type method. All the structures were solved with direct methods (SHELXS[7]) and refined by full-matrix least squares on F² using the OLEX2 software [8] which utilizes the SHELXL-2013 module [7]. No attempt was made to locate the hydrogens for disordered solvent molecules. Constraints (AFIX and EADP) and restraints (DFIX and ISOR) are used for disorder models in particular lower-rim alkyl chains.
Table S1: X-ray experimental details for 3@BrC6 - 6@BrC6

| Complex          | 3@BrC6          | 4@BrC6          | 5@BrC6          | 6@BrC6          |
|------------------|-----------------|-----------------|-----------------|-----------------|
| CCDC No:         | 1837604         | 1837605         | 1837606         | 1837607         |
| Empirical formula| C_{76}H_{96}Br_{4}N_{4}O_{12} | C_{61}H_{87}Br_{4}N_{4}O_{12} | C_{120}H_{161}Br_{8}N_{2}O_{20} | C_{60}H_{83}Br_{4}N_{2}O_{12} |
| Formula weight   | 1577.20         | 1345.95         | 2590.78         | 1329.91         |
| Temperature (K)  | 123.00(10)      | 120.00(10)      | 120.00(10)      | 120.00(10)      |
| Crystal system   | Triclinic       | Triclinic       | Triclinic       | Monoclinic      |
| Space group      | P-1             | P-1             | P.1             | P2_1/c          |
| Unit cell dimensions: a (Å) | 14.8156(5) | 13.8761(5)      | 12.6797(8)      | 21.3724(18)    |
|                  | 14.8686(6)      | 15.5316(6)      | 21.2707(13)     | 11.1431(10)    |
|                  | 17.7798(6)      | 15.6532(6)      | 24.4864(15)     | 25.7133(2)     |
|                  | 88.783(3)       | 80.377(3)       | 113.785(6)      | 90              |
|                  | 71.093(3)       | 89.829(3)       | 97.433(5)       | 97.3173(8)     |
|                  | 78.562(3)       | 69.999(3)       | 90.822(5)       | 90              |
| Volume / Å³      | 3627.8(2)       | 3120.1(2)       | 5984.9(7)       | 6073.90(9)     |
| Z                | 2               | 2               | 2               | 4               |
| Density (calculated) mg/m³ | 1.444           | 1.433           | 1.438           | 1.454           |
| Absorption Coefficient mm⁻¹ | 3.221           | 3.626           | 2.746           | 3.719           |
| F(000)           | 1632            | 1392            | 2670            | 2744            |
| Crystal size (mm³) | 0.13 x 0.12 x 0.10 x 0.08 x 0.16 x 0.10 x 0.25 x 0.17 x 0.05 0.05 0.07 0.05 | 0.16 x 0.10 x 0.25 x 0.17 x 0.07 0.05 | 0.16 x 0.10 x 0.25 x 0.17 x 0.07 0.05 | 0.16 x 0.10 x 0.25 x 0.17 x 0.07 0.05 |
| θ range for data collection (°) | 3.04 to 66.75 | 3.40 to 66.75 2.99 to 25.0 3.47 to 66.75 | 3.40 to 66.75 2.99 to 25.0 3.47 to 66.75 | 3.40 to 66.75 2.99 to 25.0 3.47 to 66.75 |
| Reflections collected | 21600 | 17730 | 53033 | 33131 |
| [R(int)]         | [0.0327]        | [0.0371]        | [0.1019]        | [0.0358]        |
| Reflections [I>2sigma(I)] | 11180 | 8958 | 11813 | 9966 |
| Data completeness (%) | 99.04 | 98.97 | 98.82 | 99.83 |
| Data/ restraints/ parameters | 127520/889 | 109534/44/712 | 2104145/1343 | 107610/705 |
| Goodness-of-fit on F² | 1.119 | 1.028 | 1.048 | 1.026 |
| Final R₁ indices [I>2sigma(I)] | R₁ = 0.0368, | R₁ = 0.0590, | R₁ = 0.0940, | R₁ = 0.0361, |
|                  | wR₂ = 0.0861 | wR₂ = 0.1547 | wR₂ = 0.2253 | wR₂ = 0.0952 |
| Final R indices [all data] | R₁ = 0.0433, | R₁ = 0.0722, | R₁ = 0.1545, | R₁ = 0.0383, |
|                  | wR₂ = 0.0983 | wR₂ = 0.1658 | wR₂ = 0.2732 | wR₂ = 0.0980 |
| Largest diff. peak/hole (e.Å⁻³) | 0.556/ | 1.675/ | 3.676/ | 1.243/ |
|                  | -0.583 | -2.546 | -0.970 | -0.552 |
| Complex         | 7@BrC6       | 8@BrC6       | MeOH+BrC6   | 11@BrC6     | 12@BrC6     |
|-----------------|--------------|--------------|-------------|-------------|-------------|
| CCDC No         | 1837608      | 1837609      | 1837610     | 1837611     | 1837612     |
| Empirical formula | C$_{130}$H$_{172}$Br$_{8}$N$_4$O$_{26}$ | C$_{60}$H$_{86}$Br$_{3}$N$_2$O$_{13}$ | C$_{60}$H$_{86}$Br$_{3}$O$_9$ | C$_{60}$H$_{86}$Br$_{4}$N$_2$O$_{13}$ | C$_{60}$H$_{86}$Br$_{4}$N$_2$O$_{13}$ |
| Formula weight  | 2845.99      | 1422.99      | 1172.74     | 1425.01     | 1392.97     |
| Temperature (K) | 120.01(10)   | 120.00(1)    | 170.0(1)    | 120.0(1)    | 120.0(1)    |
| Crystal system  | Triclinic    | Triclinic    | Triclinic   | Triclinic   | Triclinic   |
| Space group     | P-1          | P-1          | P-1         | P-1         | P-1         |
| Unit cell       | 10.7228(11)  | 11.9096(2)   | 10.501(2)   | 10.0437(3)  | 15.2700(9)  |
| Dimensions: a (Å) | 14.9600(15)  | 13.4509(2)   | 15.161(3)   | 17.0384(4)  | 18.0854(7)  |
| b (Å)           | 21.203(2)    | 20.6387(4)   | 17.053(3)   | 21.0991(7)  | 24.1760(12) |
| c (Å)           | 103.231(9)   | 91.3963(15)  | 90.81(3)    | 69.181(3)   | 75.629(4)   |
| α (°)           | 90.560(8)    | 101.1181(17) | 107.46(3)   | 76.589(3)   | 88.489(4)   |
| β (°)           | 104.941(9)   | 90.7080(14)  | 94.12(3)    | 79.976(2)   | 83.697(4)   |
| γ (°)           | 3190.4(6)    | 3242.71(11)  | 2581.3(10)  | 3266.16(18) | 6428.5(6)   |
| Volume / Å$^3$  | 1284308      | 1837610      | 1837611     | 1837612     |
| Z               | 1            | 2            | 2           | 2           | 4           |
| Dens (calculated) | 1.481        | 1.457        | 1.509       | 1.449       | 1.439       |
| Absorption Coefficient mm$^{-1}$ | 3.598        | 3.540        | 3.173       | 3.515       | 3.547       |
| F(000)          | 1468         | 1468         | 1204        | 1472        | 2872        |
| θ range for data collection (°) | 3.36 to 66.75 | 3.29 to 66.75 | 2.53 to 25.25 | 4.14 to 66.75 | 2.91 to 66.75 |
| Reflections     | 19440        | 19340        | 11117       | 18594       | 37439       |
| collected [R(int)] | [0.0989]    | [0.0227]     | [0.0404]    | [0.0407]    | [0.0618]    |
| Refinements     | 6912         | 10571        | 22555       | 9439        | 16443       |
| [I>2sigma(I)]   | 98.81        | 99.08        | 99.50       | 99.08       | 98.97       |
| Data completeness (%) | 11163/59/793 | 11378/7/793  | 9313/0/605  | 11474/11/824 | 22566/104/1550 |
| Data/ restraints/ parameters | 1.002        | 1.037        | 1.011       | 0.976       | 1.031       |
| Goodness-of-fit on F$^2$ | 0.0736       | 0.0289       | 0.0397      | 0.0468      | 0.0891,     |
| [I>2sigma(I)]  | wR$_1$ = 0.1672 | wR$_2$ = 0.0753 | wR$_3$ = 0.0803 | wR$_2$ = 0.1171 | wR$_2$ = 0.2289 |
| Final R indices | R$_1$ = 0.1217, | R$_1$ = 0.0316, | R$_1$ = 0.0627, | R$_1$ = 0.0567, | R$_1$ = 0.1142, |
| [all data]     | wR$_2$ = 0.2041 | wR$_2$ = 0.0776 | wR$_2$ = 0.1221 | wR$_2$ = 0.0882 | wR$_2$ = 0.2529 |
| Largest diff.   | 0.993/       | 0.595/       | 0.474/      | 0.990/      | 3.604/      |
| peak/hole (eÅ$^3$) | -1.669       | -0.531       | -0.487      | -0.620      | -1.372      |
III Computational study

IIIa General Information

Molecular mechanics analysis of the complexes between three \(\text{C-hexyl-2-bromoresorcinarene (BrC6)}, \ \text{C-ethyl-2-bromoresorcinarene (BrC2)}\) and \(\text{C-propyl-2-bromoresorcinarene (BrC3)}\) hosts and \(N\)-oxide \(3\) were initially carried out using Jaguar/Maestro software package [9] and OPLS-2005 force field. In order to make sure that we were adequately screening the conformer space of the complexes in these simulations, no constraints were applied on either \(N\)-oxide or acetone molecules.

The low energy conformer of \(3@\text{BrC2}, \ 3@\text{BrC3},\) and \(3@\text{BrC6}\) complexes were then optimized using the Gaussian 09 suite [10] of programs at the density functional theory (DFT) level with M062X/6-31G(d,p) [11] within the IEF-PCM solvation model [12]. All of the optimized complex geometry were confirmed by frequency calculations as minima with zero imaginary frequencies.

Single point calculations were performed on these optimized structures using long-range corrected (LRC) exchange-correlation functional with inclusion of dispersion correction, \(\omega B97X-D\) in order to obtain a more accurate treatment of stacking type interactions [13].

Structure analysis such as Molecular Electrostatic Potential (MEP) surface map, was performed using GaussView v5.0.8.4.

A topological analysis of the electron density was performed with Bader’s quantum theory of atoms in molecules (QTAIM) using the AIM2000 software [14].

Of note, the energies implemented in Table 2 are not interaction energies. We believe that calculated and predicted interaction energy for more than three components (out \(N\)-oxide, in \(N\)-oxide, as well as acetone molecule with receptor) won’t be much accurate due to basis set superposition errors” (BSSEs) and “basis set incompleteness errors” (BSIEs) and in our idea the counterpoise correction of interaction energy for removing these errors won’t be effective to completely remove the errors. The reported energies in Table 2, obtained from quantum theory of atom in molecule (QTAIM) only shows the contributions of different possible non-covalent interactions on energetic aspect and stability of the calculated structures and provide a basis to explain the presence of these attractive interactions in the systems and distinguish them from weak interactions. As mentioned in Table 2 (See manuscript), electron density \(p(r)\) and Laplacian of the electron density \(\nabla^2 p(r)\) at the BCP, is related to bond order and in turn bond strength. \(E_{(B)}\) is the energy [15] of those bonds (vary from 2.9 to 11.0 kcal/mol for H-bonds and 0.8 to 1.9 kcal/mol for other classes of non-covalent interactions) which is calculated from following equation.

Although, they have important role in energetic aspect of complex, they are not the interaction energy.

\[
E_{(B)} = 1/2V_c
\]

\[
V_c = 1/4 \nabla^2 p(r_c) - 2G_c
\]

Where \(V_c\) is the potential energy density and the kinetic energy density at the BCP.
Table S3: Isodesmic reaction schemes for comparing relative energy of the 3@BrC2, 3@BrC3 and 3@BrC6 complexes.

|   | Isodesmic Reaction Scheme          | Relative Energy (kcal/mol) |
|---|-----------------------------------|-----------------------------|
| 3@BrC2 | ![Reaction Scheme](image) | 11.3 |
| 3@BrC3 | ![Reaction Scheme](image) | 10.9 |
| 3@BrC6 | ![Reaction Scheme](image) | 0 |


**Figure S1:** The plotted molecular graph and topological properties $3@BrC6$ complex by QTAIM analysis.

**IIIb. DFT Calculated host-guest complex geometries for $3@BrC2$**

|   |   |   |   |
|---|---|---|---|
| Br | -0.21242300 | 3.75438200 | -2.27278600 |
| Br | -6.05212000 | -0.43967100 | -1.67032500 |
| Br | 0.03492100  | -3.44498000 | -3.56893200 |
| Br | 6.19317600  | -0.54492000 | -1.54851100 |
| O  | 2.20394000  | 2.49342600  | -0.42903500 |
| O  | -2.50197400 | 3.13352500  | -0.39352500 |
| H  | -2.43447900 | 4.16368700  | -0.25666000 |
| O  | -4.40569500 | 1.68856800  | -0.05979300 |
| H  | -3.95255100 | 2.32555500  | -0.12399100 |
| O  | -4.77586200 | -3.04689100 | -0.64772600 |
| H  | -5.44182900 | -2.78836000 | -1.30528500 |
| O  | -2.33428500 | -3.59395400 | -1.81293500 |
| H  | -3.16527400 | -3.63293200 | -1.30935300 |
| O  | 2.39436600  | -3.73305300 | -1.64998800 |
| H  | 2.26176900  | -3.70064600 | -2.60374300 |
| O  | 4.87404800  | -3.07835700 | -0.64998100 |
| H  | 4.17786100  | -3.67232500 | -0.97686700 |
| O  | 4.67414700  | 1.65010800  | -0.00770600 |
| H  | 5.37753600  | 1.61263400  | -0.67537100 |
| C  | -0.04870700 | 1.50024000  | 1.76357200  |
| Atoms | X-coordinates | Y-coordinates | Z-coordinates |
|-------|---------------|---------------|---------------|
| H     | -0.01433400   | 0.92094400    | 2.68191200    |
| C     | 1.15468600    | 1.83477800    | 1.14496100    |
| C     | 1.10977700    | 2.50687500    | -0.08813800   |
| C     | -0.13358000   | 2.87595300    | -0.60838800   |
| C     | -1.32207200   | 2.63704600    | 0.08463200    |
| C     | -1.28955500   | 1.89985600    | 1.27455300    |
| C     | -2.59714000   | 1.52508300    | 1.95538400    |
| H     | -3.28897600   | 2.36533100    | 1.82362100    |
| C     | -2.47728400   | 1.28144200    | 3.46820400    |
| H     | -3.46455300   | 0.43843600    | 3.82944300    |
| C     | -2.00962300   | 2.51037800    | 4.22277000    |
| H     | -2.69447900   | 3.35111300    | 4.08712600    |
| C     | -1.00846100   | 2.82291600    | 3.92610000    |
| C     | -1.97370500   | 2.30593500    | 5.31574200    |
| C     | -2.69710000   | -0.96338800   | 1.47702500    |
| H     | -1.85111700   | -1.08068300   | 2.15358900    |
| C     | -3.20818900   | -2.10242100   | 0.85651700    |
| C     | -4.26653100   | -1.93398900   | -0.04107900   |
| C     | -4.73419300   | -0.65339800   | -0.33506000   |
| C     | -4.21033700   | 0.48195200    | 0.28892500    |
| C     | -3.19544300   | 0.31798300    | 1.24638200    |
| C     | -2.59173700   | -3.47781300   | 1.09882200    |
| H     | -2.62757600   | -4.22739900   | 0.66762500    |
| C     | -2.50206800   | -3.79998400   | 2.59794100    |
| H     | -1.98794300   | -4.76157700   | 2.71483800    |
| C     | -1.88390500   | -3.04861100   | 3.10237300    |
| C     | -3.87696600   | -3.85648600   | 3.25609200    |
| H     | -4.38593900   | -2.89132400   | 3.17815800    |
| H     | -3.79145400   | -4.10965000   | 4.31566800    |
| C     | -4.50996400   | -4.61024100   | 2.77727200    |
| C     | -0.03025100   | -3.56810100   | 1.04463900    |
| H     | -0.03927700   | -3.46504000   | 2.12445500    |
| C     | 1.20238500    | -3.64159200   | 0.39754400    |
| C     | 1.19591200    | -3.69980100   | -1.00068000   |
| C     | -0.00721700   | -3.66175800   | -1.69549500   |
| C     | -1.23586300   | -3.62662800   | -1.03068000   |
| C     | -1.25444900   | -3.58219300   | 0.37084600    |
| C     | 2.54765100    | -3.52127600   | 1.10189700    |
| H     | 3.21052800    | -4.29027800   | 0.68730600    |
| C     | 2.49288600    | -3.76472800   | 2.61829900    |
| C     | 3.48846400    | -3.54686400   | 3.01936200    |
| H     | 1.81142400    | -3.04300000   | 3.08382300    |
| C     | 2.09639600    | -5.19203100   | 2.98881800    |
| H     | 2.10254500    | -5.32624100   | 4.07389000    |
| H     | 2.79880100    | -5.91189500   | 2.55730100    |
| C     | 1.09693900    | -5.44388800   | 2.62270300    |
| C     | 2.60068100    | -1.00234000   | 1.34195500    |
| H     | 1.71999900    | -1.09959800   | 1.97856400    |
| C     | 3.11593500    | 0.27190900    | 1.11467900    |
| C     | 4.21162400    | 0.39375700    | 0.25695600    |
| C     | 4.76830300    | -0.74466700   | -0.32572300   |
| C     | 4.26005300    | -2.01978000   | -0.06451700   |
| C     | 3.15443400    | -2.15668600   | 0.78928800    |
| C     | 2.49814400    | 1.49795600    | 1.77159600    |
| H     | 3.16631700    | 2.34922500    | 1.58966000    |
| C     | 2.43631400    | 1.33090000    | 3.29908900    |
|    |        |        |        |
|----|--------|--------|--------|
| H  | 1.90075500 | 2.19149300 | 3.71931600 |
| H  | 1.85032700 | 0.44023400 | 3.55221400 |
| C  | 3.82492300 | 1.22099000 | 3.92095100 |
| H  | 3.75887500 | 1.13076200 | 5.00806100 |
| H  | 4.42881700 | 2.10409700 | 3.68925200 |
| H  | 4.35433500 | 0.34244500 | 3.54026900 |
| O  | -2.31010000 | 5.58218400 | 0.19013500 |
| N  | -1.06162500 | 5.63992200 | 0.62289500 |
| C  | -0.75755000 | 5.14866400 | 1.84543400 |
| H  | -1.60700000 | 4.79695800 | 2.25456000 |
| C  | 0.55852600 | 5.08471000 | 2.25456000 |
| H  | 0.78360700 | 4.67820600 | 3.23348700 |
| C  | 1.56893700 | 5.51938500 | 1.39840600 |
| H  | 2.61023900 | 5.43370100 | 1.69206000 |
| C  | 1.23920600 | 6.05550900 | 0.15648700 |
| C  | -0.10690200 | 6.13422000 | -0.18403900 |
| H  | -0.47076500 | 6.51218400 | -0.85366000 |
| H  | 2.26814600 | 6.46081100 | -0.85366000 |
| H  | 2.06423700 | 7.45625000 | -1.25990100 |
| H  | 2.23754000 | 5.74723200 | -1.68916700 |
| H  | 3.26985900 | 6.45244500 | -0.42797500 |
| O  | 2.22061100 | 0.38337700 | -3.75850100 |
| N  | 1.16519100 | 0.18681100 | -3.04798000 |
| C  | -0.04807900 | 0.59794600 | -3.49965900 |
| H  | -0.01011200 | 1.10728000 | -4.45394100 |
| C  | -1.21025000 | 0.40586200 | -2.76781500 |
| C  | -1.12220600 | -0.25743000 | -1.54539200 |
| H  | -2.01702900 | -0.44329200 | -0.96246100 |
| C  | 0.12595200 | -0.64036200 | -1.07343100 |
| H  | 0.23506600 | -1.12631000 | -0.10885300 |
| C  | 1.25680400 | -0.42438800 | -1.83774100 |
| H  | 2.26323600 | -0.70980000 | -1.55433100 |
| C  | -2.51340300 | 0.98853600 | -3.24018800 |
| H  | -3.36525200 | 0.41891100 | -2.85655000 |
| H  | -2.57338200 | 1.02543900 | -4.33019500 |
| H  | -2.59358400 | 2.01247900 | -2.85639000 |
| O  | 0.00079600 | -1.36409500 | 3.61505300 |
| C  | -0.02512200 | -0.38217300 | 5.78702000 |
| H  | 0.74418100 | -0.49110000 | 6.55611500 |
| H  | -0.95517100 | -0.38733300 | 6.29665000 |
| H  | 0.09906400 | 0.56455800 | 5.28027000 |
| C  | 0.02305400 | -1.53941000 | 4.82371700 |
| C  | 0.10006200 | -2.92330500 | 5.41399000 |
| H  | -0.63660400 | -3.04195600 | 6.21315100 |
| H  | 1.08997600 | -3.06144000 | 5.86286200 |
| H  | -0.05424800 | -3.67937600 | 4.64181700 |
Illc. DFT Calculated host-guest complex geometries for 3@BrC3

|   |   |   |   |
|---|---|---|---|
| Br | 2.46935800 | 5.12096800 | 0.97065900 |
| Br | -4.78850000 | 2.64165900 | 1.02130200 |
| Br | -3.19170800 | -2.79471700 | -4.15443100 |
| Br | 4.00981400 | 0.13405400 | -4.25348700 |
| Br | 2.26120200 | -2.35097000 | -3.78519200 |
| Br | 2.58497200 | -1.84472500 | -4.54671000 |
| Br | -0.29597900 | -3.46035900 | -3.82644700 |
| Br | 0.63444000 | -3.17412200 | -3.84563900 |
| Br | -4.18807800 | -2.45270700 | -1.37424500 |
| Br | -4.41656500 | -1.83394000 | -0.66100800 |
| Br | -4.62905300 | -0.40843400 | 0.35844700 |
| H | -5.18308400 | 0.27483300 | -0.10778900 |
| H | -2.85054000 | 3.09492900 | 2.71310700 |
| H | -3.01464500 | 3.62776100 | 2.35868000 |
| H | 0.31320500 | 4.06857900 | 2.71914100 |
| H | -0.60073900 | 3.74273300 | 2.81913600 |
| O | 4.34616500 | 2.69882000 | 0.58654800 |
| O | 4.27091200 | 3.51895300 | 0.04182500 |
| O | 4.72000000 | 1.02320700 | -1.51732100 |
| O | 4.72116500 | -1.69184300 | -0.69184300 |
| C | -1.56598300 | -0.43369700 | 2.29880700 |
| C | -0.81444000 | -1.15540100 | 2.60954600 |
| C | -1.42689400 | 0.88974300 | 2.71183600 |
| C | -2.40226500 | 1.79907400 | 2.30030100 |
| C | -3.47106900 | 1.37796100 | 1.50583900 |
| C | -3.59564000 | 0.04276000 | 1.10942300 |
| C | -2.63685300 | -0.88939100 | 1.53510000 |
| C | -2.78934600 | -2.36631300 | 1.18111100 |
| H | -3.86143800 | -2.56054000 | 1.06486500 |
| C | -2.11204300 | -2.67975600 | -0.14637100 |
| C | -2.85036600 | -2.66479000 | -1.34014700 |
| C | -2.19841900 | -2.90920400 | -2.55162500 |
| C | -0.83438900 | -3.21228300 | -2.60786800 |
| C | -0.09661300 | -3.26859600 | -1.41364500 |
| C | -0.75219100 | -2.97618900 | -0.21836600 |
| C | -0.17097100 | -2.97604900 | 0.70204100 |
| C | 1.38886400 | -3.60470200 | -1.42753300 |
| C | 1.60139200 | -4.13155300 | -2.36518800 |
| C | 2.24435000 | -2.34117200 | -1.41700400 |
| C | 2.65969200 | -1.76495600 | -2.62074900 |
| C | 3.46408900 | -0.62655600 | -2.60812700 |
| C | 3.89481500 | -0.04063000 | -1.41288100 |
| C | 3.48370200 | -0.60450100 | -0.19365700 |
| C | 2.65061200 | -1.72313100 | -0.23670400 |
| C | 2.31946600 | -2.14770500 | 0.70777900 |
| C | 3.95460600 | -0.02139500 | 1.13828700 |
| H | 4.92543100 | 0.45616100 | 0.96409100 |
| C | 2.99131700 | 1.05706200 | 1.62329000 |
| C | 1.84774700 | 0.75337500 | 2.35991500 |
| H | 1.64823200 | -0.28766000 | 2.60710500 |
| C | 0.94126000 | 1.72247800 | 2.79290200 |
| C | 1.15631500 | 3.05485300 | 2.41263700 |
| C | 2.28835100 | 3.37197500 | 1.65382400 |
| C | 3.21715000 | 2.39792900 | 1.28591200 |
| C | -0.28216600 | 1.34003300 | 3.61263000 |
C  4.20835300  4.02432400  -3.91334600
H  5.05181400  4.70954600  -3.98532000
H  3.35406800  3.05092300  -4.33602400
C  3.82707100  3.80358000  -2.47744300
C  2.53867900  3.06524500  -2.22397400
H  2.41230400  2.25728300  -2.95072700
H  1.70602600  3.76794900  -2.34421500
H  2.51116200  2.66082500  -1.21143300
O  1.09660500  -2.61170400  2.72781800
C  2.08409200  -3.00216900  4.85994800
H  2.49677000  -1.99781300  4.75009700
H  2.87059400  -3.71569300  5.12050800
H  1.36162700  -3.00648900  5.68442500
C  1.37404100  -3.42236500  3.60061800
C  1.02192300  -4.87887500  3.45637000
H  1.93968600  -5.43504600  3.23216200
H  0.30848700  -5.02312200  2.64322000
H  0.62528300  -5.27930400  4.39301500
O  -6.27427200  1.00906900  -1.15044000
C  -6.72960400  2.25520600  -3.11687600
H  -7.19961000  1.85886100  -4.02106100
H  -5.95308000  2.96292800  -3.42246800
H  -7.46684500  2.75944100  -2.49291100
C  -6.05882000  1.15592200  -2.34648000
C  -5.16611200  0.22642100  -3.12192900
H  -4.57535300  -0.40302700  -2.45688800
H  -4.52103800  0.80069900  -3.78976400
H  -5.79502800  -0.42682500  -3.73757500
IIIc. DFT Calculated host-guest complex geometries for 3@BrC6

| Atom | x     | y     | z     |
|------|-------|-------|-------|
| Br   | -6.03679900 | -1.83405900 | -0.14046300 |
| Br   | -4.15722300 | 5.56463300  | -0.97836900  |
| Br   | -0.51799800 | 2.87327800  | 5.32892700   |
| Br   | -1.03029900 | -4.93054200 | 4.54504800   |
| O    | -4.89634600 | 0.48223500  | -1.81862500  |
| H    | -5.73665900 | 0.25688300  | -1.39125000  |
| O    | -3.66637200 | -3.73981700 | 0.11814300   |
| H    | -4.40259900 | -4.22031700 | -0.43554400  |
| O    | -4.15611100 | 3.13027100  | -2.86499900  |
| H    | -4.70291400 | 3.91282000  | -2.90553300  |
| O    | -1.50883300 | 5.41803900  | 0.35753100   |
| H    | -0.96108500 | 5.18123700  | 1.12649600   |
| O    | -0.42880200 | 4.22952200  | -0.54554500  |
| H    | -0.77602300 | 4.42716100  | 3.47582400   |
| O    | 0.47220500  | 0.03745600  | 4.72671300   |
| H    | 0.95233800  | -2.67445800 | 4.65181900   |
| C    | 0.13200100  | 1.53091000  | 5.87790900   |
| H    | 0.13938000  | 0.49019000  | 6.23401500   |
| C    | 1.46289600  | 1.83078500  | 5.18669700   |
| H    | 1.53170700  | 2.90927600  | 4.98234000   |
| H    | 1.50720700  | 1.33723800  | 4.20345900   |
| C    | 2.66151500  | 1.39240900  | 6.02412200   |
| H    | 2.59046700  | 0.31209300  | 6.20639600   |
| H    | 2.60969600  | 1.87400100  | 7.00902800   |
| C    | 3.99303800  | 1.71614600  | 5.35802100   |
| H    | 4.84240000  | 1.34085700  | 5.93622100   |
| H    | 4.04271900  | 1.26759600  | 4.35902300   |
| H    | 4.11214600  | 2.79853700  | 5.24237900   |
| C    | -2.14720500 | 2.11296200  | -2.11242000  |
| C    | -3.07053800 | 3.16783800  | -2.06041100  |
| C    | -2.83505100 | 4.22498200  | -1.17575800  |
| C    | -1.65832600 | 4.32229300  | -0.42702100  |
| C    | -0.68762400 | 3.31728700  | -0.54554500  |
| C    | -0.98436000 | 2.21698700  | -1.35145800  |
|    |               |                |                |
|----|---------------|----------------|----------------|
| H  | -0.25957500   | 1.40277600    | -1.39911100   |
| C  | 0.65952400    | 3.42358800    | 0.15946700    |
| H  | 0.78731500    | 4.44914900    | 0.52073800    |
| C  | 1.83652300    | 3.18175900    | -0.79548400   |
| H  | 2.74583000    | 3.19901000    | -0.18270500   |
| C  | 1.77188400    | 2.18914900    | -1.25964400   |
| H  | 1.94972000    | 4.24849300    | -1.88157900   |
| H  | 1.16214600    | 4.10429500    | -2.63443300   |
| H  | 1.78107100    | 5.24087100    | -1.44013700   |
| C  | 3.32374900    | 4.23277500    | -2.54766800   |
| H  | 4.07957000    | 4.43202700    | -1.77600900   |
| H  | 3.53405700    | 3.22340100    | -2.93123400   |
| C  | 3.46375700    | 5.24435200    | -3.68448200   |
| H  | 3.07506000    | 6.21768300    | -3.35337200   |
| H  | 2.82515500    | 4.92449000    | -4.52019000   |
| C  | 4.90380100    | 5.42819900    | -4.17493400   |
| H  | 5.34858800    | 4.44503900    | -4.37642800   |
| H  | 4.89365300    | 5.96377800    | -5.13087800   |
| C  | 5.78608500    | 6.19187200    | -3.18786300   |
| H  | 6.79762600    | 6.32223500    | -3.58169700   |
| H  | 5.86952900    | 5.67081600    | -2.22977000   |
| C  | 5.37156600    | 7.18569600    | -2.99017700   |
| C  | 0.70785300    | 2.53076600    | 1.38430200    |
| C  | 0.12352400    | 2.98256700    | 2.56962200    |
| C  | 0.12897000    | 2.16949700    | 3.70023900    |
| C  | 0.64250100    | 0.87170600    | 3.66596400    |
| C  | 1.31908200    | 0.44686000    | 2.51760600    |
| C  | 1.32599700    | 1.28137800    | 1.39876200    |
| H  | 1.78859600    | 0.92436900    | 0.48194500    |
| C  | 1.96561900    | -0.92586500   | 2.52778600    |
| H  | 2.41882700    | -1.06175700   | 3.51761600    |
| C  | 3.10143400    | -1.06790100   | 1.50401100    |
| H  | 2.69413700    | -1.06587400   | 0.48390100    |
| H  | 3.74773100    | -0.18433800   | 1.58787100    |
| C  | 3.94026000    | -2.32729800   | 1.71447900    |
| H  | 4.24112300    | -2.38850000   | 2.76997700    |
| H  | 3.33190900    | -3.21994400   | 1.51125400    |
| C  | 5.19253200    | -2.34209800   | 0.84015700    |
| H  | 4.89691200    | -2.33031700   | -0.21992900   |
| H  | 5.75618000    | -1.41282400   | 1.01274800    |
| C  | 6.10821500    | -3.53537200   | 1.10400700    |
| H  | 5.58864800    | -4.46551800   | 0.83274500    |
| H  | 6.31109400    | -3.60199400   | 2.18209300    |
| C  | 7.43644200    | -3.46284700   | 0.35480000    |
| H  | 7.24321300    | -3.39497900   | -0.72319900   |
| H  | 7.95815500    | -2.53862000   | 0.63533300    |
| C  | 8.33006100    | -4.66870400   | 0.63206200    |
| H  | 9.28097800    | -4.59797100   | 0.09763600    |
| H  | 7.83696600    | -5.59580400   | 0.32104000    |
| H  | 8.54927100    | -4.75321300   | 1.70127300    |
| C  | 0.91352200    | -2.00451000   | 2.33398100    |
| C  | 0.48780900    | -2.82761800   | 3.38609400    |
| C  | -0.44653200   | -3.83539100   | 3.12306700    |
| C  | -0.96594100   | -4.04616000   | 1.83955900    |
| C  | -0.53810400   | -3.22078100   | 0.78733300    |
| H  | 0.35748600    | -2.19524200   | 1.07341300    |
| H  | 0.66013200    | -1.52556700   | 0.26737500    |
S15
C  -2.06908200  -0.25500200  2.68593300
H  -1.56803300  -1.18988400  2.90731000
C  -3.06411300  1.65770500  3.62668900
H  -3.28571200  2.15708000  4.56060200
C  -3.35267800  2.17836000  2.37212200
C  -3.98747000  3.54193300  2.22401300
H  -4.68255300  3.54729800  1.38062500
H  -3.21626900  4.28405700  2.01402500
H  -4.51968700  3.83130800  3.12954300
O  2.99989100  4.73370000  1.86683400
N  3.56991900  3.71862700  2.41099700
C  4.69208080  3.08613700  1.97796230
H  4.87901000  3.53011800  0.83519000
C  5.19504300  1.97631100  2.34444900
H  6.00639000  1.49730400  1.80742100
C  4.75247000  1.48669100  3.56807100
H  5.20492900  0.60402000  4.00903300
C  3.71031600  2.14648700  4.22294400
C  3.15006700  3.26219000  3.61963800
H  2.33151700  3.82663500  4.04819900
C  3.19431200  1.68332300  5.55864800
H  3.12605900  0.59295600  5.58653200
H  2.20129100  2.08903000  5.76322300
H  3.86910800  1.99989300  6.35963400
IV Solution studies

A 10 mM stock solution of the C-hexyl-2-methylresorcinarene BrC6 host and 20 mM of the guests were prepared in the respective solvents (acetone-d₆, CD₃OD/CDCl₃ 1:1 v/v and CD₃OD/DMSO-d₆ 9:1 v/v). For the pure host, aliquot 300 μL of the 10 mM stock solution and diluted with the NMR solvent to final volume 450 μL with a concentration 6.6 mM. For the pure guest, aliquot 150 μL of the 20 mM stock solution and diluted to a final volume 450 μL with a concentration 6.6 mM. For the 1:1 host–guest, aliquot 300 μL of host and 150 μL of guest were mixed to final volume 450 μL to give a concentration 6.6 mM.

Figure S2: An expansion of the ¹H NMR (6.6 mM at 298 K, 500 MHz) of BrC6 complexes with 1. Spectra are produced from BrC6, 1 and an equimolar mixture of BrC6 and 1 in: (a) (CD₃)₂O, (b) CD₃OD/CDCl₃ 1:1 v/v, and (c) CD₃OD/DMSO-d₆ 9:1 v/v. Dashed lines highlight the observed shift changes of the resonances, labels are in ppm.
Figure S3: An expansion of the $^1$H NMR (6.6 mM at 298 K, 500 MHz) of BrC6 complexes with 2. Spectra are produced from BrC6, 2 and an equimolar mixture of BrC6 and 2 in: (a) (CD$_3$)$_2$O, (b) CD$_3$OD/CDCl$_3$ 1:1 v/v, and (c) CD$_3$OD/DMSO-$d_6$ 9:1 v/v. Dashed lines highlight the observed shift changes of the resonances, labels are in ppm.
Figure S4: An expansion of the $^1$H NMR (6.6 mM at 298 K, 500 MHz) of BrC6 complexes with 4. Spectra are produced from BrC6, 4 and an equimolar mixture of BrC6 and 4 in: (a) (CD$_3$)$_2$O, (b) CD$_3$OD/CDC$_3$ 1:1 v/v, and (c) CD$_3$OD/DMSO-d$_6$ 9:1 v/v. Dashed lines highlight the observed shift changes of the resonances, labels are in ppm.
Figure S5: An expansion of the $^1$H NMR (6.6 mM at 298 K, 500 MHz) of BrC6 complexes with 5. Spectra are produced from BrC6, 5 and an equimolar mixture of BrC6 and 5 in: (a) (CD$_3$)$_2$O, (b) CD$_3$OD/CDC$_3$ 1:1 v/v, and (c) CD$_3$OD/DMSO-$d_6$ 9:1 v/v. Dashed lines highlight the observed shift changes of the resonances, labels are in ppm.
Figure S6: An expansion of the $^1$H NMR (6.6 mM at 298 K, 500 MHz) of BrC6 complexes with 6. Spectra are produced from BrC6, 6 and an equimolar mixture of BrC6 and 6 in: (a) (CD$_3$)$_2$O, (b) CD$_3$OD/CDC1$_3$ 1:1 v/v, and (c) CD$_3$OD/DMSO-$d_6$ 9:1 v/v. Dashed lines highlight the observed shift changes of the resonances, labels are in ppm.
Figure S7: An expansion of the $^1$H NMR (6.6 mM at 298 K, 500 MHz) of BrC6 complexes with 7. Spectra are produced from BrC6, 7 and an equimolar mixture of BrC6 and 7 in: (a) (CD$_3$)$_2$O, (b) CD$_3$OD/CDC$_3$ 1:1 v/v, and (c) CD$_3$OD/DMSO-d$_6$ 9:1 v/v. Dashed lines highlight the observed shift changes of the resonances, labels are in ppm.
Figure S8: An expansion of the $^1$H NMR (6.6 mM at 298 K, 500 MHz) of BrC6 complexes with 8. Spectra are produced from BrC6, 8 and an equimolar mixture of BrC6 and 8 in: (a) (CD$_3$)$_2$O, (b) CD$_3$OD/CDCl$_3$ 1:1 v/v, and (c) CD$_3$OD/DMSO-δ$_6$ 9:1 v/v. Dashed lines highlight the observed shift changes of the resonances, labels are in ppm.
Figure S9: An expansion of the $^1$H NMR (6.6 mM at 298 K, 500 MHz) of BrC6 complexes with 9. Spectra are produced from BrC6, 9 and an equimolar mixture of BrC6 and 9 in: (a) (CD$_3$)$_2$O, (b) CD$_3$OD/CDC$_3$ 1:1 v/v, and (c) CD$_3$OD/DMSO-$d_6$ 9:1 v/v. Dashed lines highlight the observed shift changes of the resonances, labels are in ppm.
Figure S10: An expansion of the $^1$H NMR (6.6 mM at 298 K, 500 MHz) of BrC6 complexes with 10. Spectra are produced from BrC6, 10 and an equimolar mixture of BrC6 and 10 in: (a) (CD$_3$)$_2$O, (b) CD$_3$OD/CDCl$_3$ 1:1 v/v, and (c) CD$_3$OD/DMSO-d$_6$ 9:1 v/v. Dashed lines highlight the observed shift changes of the resonances, labels are in ppm.
Figure S11: An expansion of the $^1$H NMR (6.6 mM at 298 K, 500 MHz) of BrC6 complexes with 11. Spectra are produced from BrC6, 11 and an equimolar mixture of BrC6 and 11 in: (a) (CD$_3$)$_2$O, (b) CD$_3$OD/CDCl$_3$ 1:1 v/v, and (c) CD$_3$OD/DMSO-$d_6$ 9:1 v/v. Dashed lines highlight the observed shift changes of the resonances, labels are in ppm.
Figure S12: An expansion of the $^1$H NMR (6.6 mM at 298 K, 500 MHz) of BrC6 complexes with 12. Spectra are produced from BrC6, 12 and an equimolar mixture of BrC6 and 12 in: (a) (CD$_3$)$_2$O, (b) CD$_3$OD/CDCl$_3$ 1:1 v/v, and (c) CD$_3$OD/DMSO-d$_6$ 9:1 v/v. Dashed lines highlight the observed shift changes of the resonances, labels are in ppm. No analysis of between BrC6 and 12 in acetone-d$_6$ due to limited solubility of 12 at 6.6 mM.
V References

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