Heteronanostructured Co@carbon nanotubes-graphene ternary hybrids: synthesis, electromagnetic and excellent microwave absorption properties

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In order to explore high efficiency microwave absorption materials, heteronanostructured Co@carbon nanotubes-graphene (Co@CNTs-G) ternary hybrids were designed and produced through catalytic decomposition of acetylene at the designed temperature (400, 450, 500 and 550 °C) over Co3O4/reduced graphene oxide (Co3O4/RGO). By regulating the reaction temperatures, different CNT contents of Co@CNTs-G ternary hybrids could be synthesized. The investigations indicated that the as-prepared heteronanostructured Co@CNTs-G ternary hybrids exhibited excellent microwave absorption properties, and their electromagnetic and microwave absorption properties could be tuned by the CNT content. The minimum reflection loss (RL) value reached approximately −65.6, −58.1, −41.1 and −47.5 dB for the ternary hybrids synthesized at 400, 450, 500 and 550 °C, respectively. And RL values below −20 dB (99% of electromagnetic wave attenuation) could be obtained over the as-prepared Co@CNTs-G ternary hybrids in the large frequency range. Moreover, based on the obtained results, the possible enhanced microwave absorption mechanisms were discussed in details. Therefore, a simple approach was proposed to explore the high performance microwave absorbing materials as well as to expand the application field of graphene-based materials.

Microwave absorbing materials (MAMs) have attracted more and more attention over the world due to the expanded electromagnetic (EM) interference problems to electronic communication devices and invisible harm to biological system1–4. According to EM energy conversion principle, the reflection coefficient of an ideal absorber should be zero when it fulfills the optimal EM impedance matching condition5–7. However, single component dielectric loss material such as ZnO or magnetic loss material such as Fe3O4 is very difficult to meet this condition due to the mismatch in the values of complex permittivity ($\varepsilon_r = \varepsilon' - j\varepsilon''$) and complex permeability ($\mu_r = \mu' - j\mu''$)8–10. Based on the impedance matching strategy, one of the effective ways to solve the problem is to couple dielectric materials with magnetic nanoparticles. Among these hybrids, heteronanostructured hybrids have been proved to exhibit strong EM wave absorption properties because of their interfacial and synergistic effects. Therefore, different categories of heteronanostructured hybrids were developed to further improve their EM wave absorbing capabilities, such as Fe@ZnO11, Fe3O4@SiO212, Fe3O4@SnO213, FeCo@ZnO14, Ni@TiO215, and so on16–19. Generally, the previously reported results indicated that the microwave absorption (MA) capability of MAMs was mainly determined by the $\varepsilon_r$, $\mu_r$, EM impedance matching and microstructure. However, these hybrids still suffer from some problems such as high density, large thickness of absorber, optimal RL usually above −40 dB, and so on. Currently, high performance MAMs with light weight, thin thickness, good chemical stability, strong absorption ability and wide absorption frequency are highly desired.

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Graphene-based materials (graphene, graphene oxide, etc), have become the current focus because of their outstanding physical and chemical properties such as the excellent thermal and electronic conductivity, huge specific surface area, and strong mechanical strength. The unique and fascinating properties make graphene and graphene-based hybrids become the promising candidates in the fields of gas sensing, field-effect transistors, supercapacitors, and MAMs. Unfortunately, graphene-based materials alone cannot achieve ideal MA properties because of the mismatch between relatively high dielectric loss and low magnetic loss. The incorporation of magnetic materials into graphene-based materials is an effective strategy to obtain excellent MA properties, and previous theoretical studies indicate that the interfacial electronic interaction between metal and graphene can make graphene show some novel magnetic and electric properties. Therefore, different categories of magnetic particles-graphene based hybrids have been developed to improve MA properties in the recent years. Herein, we design and synthesize a new Co@carbon nanotubes-graphene (Co@CNTs-G) ternary hybrid. The aim of designing these hybrids has mainly three points: firstly, Co nanoparticles and CNTs/graphene can be used as magnetic and dielectric loss materials, respectively. By controlling their compositions, a good impedance matching may be obtained on these hybrids. Secondly, both the tube and layer structures of CNTs and graphene are very beneficial to improve MA performance of hybrids in theory. Finally, the heteronanostructure of hybrid can provide the synergetic effect of Co, CNTs and graphene, which should enhance MA capability greatly.

Results

Figure 1 presents the XRD pattern, Raman spectrum and TEM images of Co$_3$O$_4$/reduced graphene oxide (Co$_3$O$_4$/RGO). As shown in Fig. 1a, the diffraction peaks (as indicated by the symbols) at 31.3, 36.8, 44.8, 59.3 and 65.2° can be assigned to (220), (311), (400), (511) and (440) crystal planes of the face-centered cubic phase Co$_3$O$_4$ (JCPDS: 78–1970), respectively. And the broad diffraction peak at ca. 21.9° is attributed to the graphite-like structure, which should be indexed to RGO. In order to obtain more information about RGO, the Raman spectrum of Co$_3$O$_4$/RGO is presented in Fig. 1b, two strong peaks centered at ca. 1348 (D band) and 1598 cm$^{-1}$ (G band) can be observed clearly. As we know that the D band is related to the defect or disorder in the graphitic structure and G band is indicative of high crystallinity graphitic layer. The peaks appeared in the range of 2500–3000 cm$^{-1}$ are characteristic Raman signal of RGO. Additionally, the A$_{1g}$ peak of Co$_3$O$_4$ can also be seen at ca. 670 cm$^{-1}$. The morphology of Co$_3$O$_4$/RGO was investigated by TEM. As shown in Fig. 1c, large quantities of Co$_3$O$_4$ nanoparticles are anchored on RGO surface without serious aggregation. The high-resolution TEM (HRTEM) image (as shown in Fig. 1d) exhibits a clear crystal lattice with an interplanar spacing of 0.24 nm, which corresponds to the (311) plane of Co$_3$O$_4$. The energy dispersive X-ray spectroscopy (EDS) result (as shown in the inset of Fig. 1) indicates that the catalyst precursor (Co$_3$O$_4$/RGO) contains C, Co and O, and their atomic percentage is ca. 69.3:28, which implies that RGO should contain a certain amount of oxygen. Generally, all the obtained results indicate that the catalyst precursor is Co$_3$O$_4$/RGO.

Based on our previous works, the method of chemical vapor deposition (CVD) was used to synthesize Co@CNTs-G ternary hybrids via catalytic decomposition of acetylene at the selected temperature over Co$_3$O$_4$/RGO. As depicted in experimental section, compared to the amount of Co$_3$O$_4$/RGO, much higher quantities of...
samples could be collected after the designed CVD processes. In order to investigate their phases and microstructures, the obtained samples were characterized by XRD, Raman, FT-IR, SEM and TEM. Figure 2 shows the XRD patterns, Raman and FT-IR spectra of the obtained Co$_3$O$_4$/RGO and Co@CNTs-G. As shown in Fig. 2a, the diffraction peaks of the as-prepared Co@CNTs-G ternary hybrids (G400, G450, G500 and G550) located at ca. 41.8, 44.4, 47.3 and 51.6° can be assigned to the phase of Co (JCPDS: 01–1254). And the diffraction peak centered at ca. 26.2° can be assigned to the (002) crystal plane of hexagonal phase graphite (JCPDS: 75–1621). The comparison XRD results (as shown in Figs 1 and 2) indicate that the sample undergoes a transform process from RGO and Co$_3$O$_4$ to graphene and Co, respectively, through the annealing treatment of Co$_3$O$_4$/RGO in acetylene. Moreover, one can find that the relative XRD peak intensity ratio of Co and graphene is decreasing with the increased temperature, which should be related to the different Co contents of the as-prepared Co@CNTs-G ternary hybrids. In our experiment, the obtained experimental results indicated that over the same amount of Co$_3$O$_4$/RGO (0.02 g), much larger quantities of Co@CNTs-G could be collected at the higher temperature, which is consistent with the different intensity ratios of Co and graphene in XRD patterns. Figure 2b presents the Raman spectra of the obtained Co@CNTs-G ternary hybrids. Similar to that of Co$_3$O$_4$/RGO (as indicated in Fig. 2b), four peaks can be observed clearly, which are indexed to D, G, 2D (intrinsic peak of graphene) and D$^+$D$^{\prime}$ bands, respectively. Combined with the XRD results, the existence of 2D and D$^+$D$^{\prime}$ bands indicates the presence of graphene. Comparison of the obtained Raman results, it can be seen that the 2D and D$^+$D$^{\prime}$ bands almost cannot be observed over the obtained G550, which may be related to the low content of graphene in the obtained hybrid. In order to detect the degree of removing the oxygen-containing functional group$^{46-48}$, the FT-IR spectra of Co$_3$O$_4$/RGO, G500 and G550 are shown in Fig. 2c and d. As shown in Fig. 2c, the broad absorption band observed at 3373 cm$^{-1}$ corresponds to the stretching vibration of intermolecular hydrogen bond O-H, and the characteristic peaks appear at 1560 and 1218 cm$^{-1}$ are due to the stretching vibration of C=C and epoxy C-O, respectively. The detection of oxygen-containing functional group over Co$_3$O$_4$/RGO is consistent with the EDS result (as shown in Fig. 1). And no IR signals such as –OH and C-O are detected over the as-prepared Co@CNTs-G (as shown in Fig. 2d). These results point out that the chemical reduction of Co$_3$O$_4$/RGO with C$_2$H$_2$ significantly reduces the oxygen species.

Figure 3 presents the SEM and TEM images of the obtained G400. As shown in Fig. 3a and b, flexible two-dimensional graphene sheet and CNTs can be found in large scale. Moreover, one can observe clearly that Co nanoparticles distribute throughout the obtained G400. As shown in the inset of Fig. 3, the EDS result (obtained from the area as indicated by the red square in Fig. 3b) shows that only C and Co can be detected. The disappearance of O over the as-prepared hybrid is in accordance with the obtained FT-IR results (as shown in Fig. 2d), implying that RGO is reduced to graphene during the CVD process. As shown in Fig. 3c,d, the TEM
investigation indicates that Co nanoparticles are encapsulated tightly by CNTs and graphene sheet. Moreover, the AFM images of Co$_3$O$_4$/RGO and G400 (as shown in Figure S1) also demonstrate that their layered morphology, which is consistent with their TEM and SEM observations. Because the thicknesses of RGO and graphene are within a large range (as shown in Figures S1b and d), it is very difficult to determine the variation in thickness during the CVD process. Moreover, similar to that of G400, the SEM and TEM investigations (as shown in Figure S2) shows that two-dimensional graphene sheet and CNTs can also be found in the obtained G450. Compared to that of G400, one can find that much more CNTs exist in the obtained G450. In the experimental section, using the same amount of Co$_3$O$_4$/RGO as catalyst precursor, larger quantities of G450 could be collected than G400 after cooling to room temperature (RT). Based on the aforementioned results, it can be seen that the increase in weights of the collected samples should be ascribed to the growth of CNTs. Therefore, higher content of CNTs can be obtained in G450 than that of G400, which means that CNTs can be observed more easily in G450. Generally, the obtained results indicate that heteronanostructured Co/CNTs-G ternary hybrids can be synthesized in large-scale through the catalytic decomposition of acetylene over Co$_3$O$_4$/RGO at the designed temperature.

Figure 4 displays the SEM and TEM images of the as-prepared G500. As shown in Fig. 4a and b, the flexible two-dimensional graphene sheet and CNTs can be observed clearly. The SEM observation indicates that the obtained G500 is also heteronanostructured Co/CNTs-G ternary hybrid. As shown in Fig. 4c, one can observed clearly that the hollow tubular structure of CNTs. Additionally, Co nanoparticles encapsulated at the top of CNTs and inlaid into graphene sheet can also be seen clearly (as shown in Fig. 4c and d). Compared to those of G400 and G450, the SEM and TEM investigations reveal that much higher CNT content can be obtained in G500. Figure 5 presents the microstructure of G550. As shown in Fig. 5a and b, similar to those of G400, G450 and G500, two-dimensional graphene sheet and CNTs can be observed obviously in the obtained sample. As shown in the inset of Fig. 5, the EDS result, which is obtained from the area as indicated by the red square in Fig. 5b, indicates that only C and Co can be detected over G550. Similar to that of G400, the disappearance of O over the as-prepared hybrid further confirms that RGO is reduced further during the CVD process. And the tube structure of CNTs, Co nanoparticles enwrapped by CNTs and graphene sheet can also be seen evidently (as shown in Fig. 5c and d). Differently from those of G400, G450 and G500, the SEM and TEM investigations show that much larger quantities of CNTs can be produced and graphene is seldom seen in the obtained G550. In generally, all the results show that the as-prepared G550 is also heteronanostructured Co/CNTs-G ternary hybrid. Combined with the obtained results, it can be seen clearly that: (1) through the catalytic decomposition of acetylene at 400–550°C over Co$_3$O$_4$/RGO, CNTs can be synthesized in large-scale; (2) all the obtained samples are heteronanostructured Co/CNTs-G ternary hybrids; (3) With the rise of temperature, Co/CNTs-G hybrids with much higher CNT content can be obtained. Therefore, a simple and efficient scheme is proposed to synthesize heteronanostructured Co/CNTs-G ternary hybrids.
Figure 6 displays the M-H curves of the obtained samples acquired at 300 K. The comparison results reveal that: (1) all the obtained Co/CNTs-G ternary hybrids exhibit typical ferromagnetic properties at RT; (2) as shown in Table 1, one can find that the experiment shows an excellent reproducibility and the saturation magnetization ($M_s$) value of the obtained hybrids decreases with the increase of temperature. Based on the aforementioned results, their good ferromagnetic properties at RT should be ascribed to the Co nanoparticles. As revealed in Table 1, over the same amount of Co$_3$O$_4$/RGO, much larger quantities of CNTs can be synthesized with the increasing temperature. In other words, the Co content in the obtained hybrids decreases with the rise of temperature. Therefore, the obtained $M_s$ value of hybrids should be as follows: $G400 > G450 > G500 > G550$, which is consistent with the obtained $M_s$ results (as shown in Table 1). As shown in Fig. 6a and b, the $M_s$ values of

Figure 4. (a,b) SEM, and (c,d) TEM images of G500.

Figure 5. (a) SEM (inset: EDS result), and (b–d) TEM images of G550.
G400 and G450 are 23.9 and 21.2 emu/g, respectively. Compared to those of magnetic graphene-based hybrids reported before49–51, the as-synthesized Co@CNTs-G ternary hybrids display an enhanced magnetic property due to high content of magnetic nanoparticle. Based on the Co content in the obtained hybrids and the \( M_s \) value of Co at 300 K, the theoretical \( M_s \) values of G400, G450, G500 and G550 should be 26.1, 14.2, 10.4 and 4.1 emu/g, respectively. The disparity between the calculated and experimental results may be related to nanosize, uneven distribution and different crystallinities of magnetic nanoparticles52,53. Moreover, it is observed that the obtained Co@CNTs-G ternary hybrids exhibit no changes in XRD patterns and magnetic properties after being kept in air for one month, which confirms further that the ferromagnetic Co nanoparticles are encapsulated tightly in CNTs and graphene sheets. In generally, because of Co nanoparticles tightly wrapped by the graphitic layers, the as-prepared hybrids exhibit good stabilities and magnetic properties at RT, and the \( M_s \) and \( H_c \) values can be tuned by the temperature, which may expand their potential application in magnetic date storage and human tumor therapy effectively. Although the introduction of CNTs decreases the \( M_s \) value of the obtained hybrids, their complex permittivity and chemical stability of Co nanoparticle can be enhanced evidently, which is conducive to obtain a good impedance matching and improve their MA capability.

According to the transmission line theory, the values of reflection loss (RL) and attenuation constant (\( \alpha \)) were calculated by the following equations54,55:

\[
\text{RL} = 20 \log_{10} \left( \frac{1}{\sqrt{\varepsilon_r}} \right)
\]

\[
\alpha = 2 \pi f \sqrt{\mu_r \varepsilon_r}
\]

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Table 1. Effects of pyrolysis temperature on products.
Based on the equations (1) and (2), the RL values of as-synthesized Co/CNTs-G ternary hybrids are obtained and the results are shown in Fig. 7. It can be seen clearly that: (1) the minimum RL value of the obtained hybrids moves towards the lower frequency region with the increasing thickness; (2) the minimum RL values for G400 and G450 are ca. −65.6 dB at 12.4 GHz with the matching thickness of 2.19 mm and −58.1 dB at 11.8 GHz with a matching thickness of 1.98 mm, respectively; (3) the optimum RL values for G450 and G500 are ca. −41.1 and −47.5 dB at 10.8 and 8.2 GHz with a matching thickness of 2.03 and 2.65 mm, respectively; (4) RL values below −20 dB (99% of EM wave attenuation) for G400, G450, G500 and G550 can be obtained in the frequency range of 2.5–18.0, 5.0–15.8, 5.2–12.6 and 2.6–11.0 GHz, respectively; (5) RL values below −10 dB (90% of EM wave attenuation) for G400, G450, G500 and G550 can be observed in the frequency range of 2.0–18.0, 2.4–18.0, 2.6–18.0 and 2.0–18.0 GHz. Generally, as shown in Table 2, the MA performance of the obtained hybrids is as follows: G400 > G450 > G550 > G500. In addition, due to the synergetic effect of Co nanoparticles, CNTs and graphene, the as-prepared Co/CNTs-G ternary hybrids show the superior absorption properties among other similar hybrids (as illustrated in Table 2).

**Discussion**

Based on the aim of experiments and the results obtained, we find that the composition of as-prepared hybrids can be regulated by the pyrolysis temperature. And the formation of heteronanostructured Co@CNTs-G ternary hybrids may be explained by the following reactions:

\[ 4C_{3}H_{2} + 5Co_{2}O_{4} = 8CO_{2} + 15Co + 4H_{2}O \]  

(4)

\[ C_{2}H_{2}^{Co} = 2C + H_{2} \]  

(5)
Among these reactions, reaction (4) is a complete reduction reaction between C_2H_2 and Co_3O_4, which can occur at the designed temperature (400–550 °C). The elemental of Co can form through the reduction of Co_3O_4 by C_2H_2. It is well known that the transition metals (such as Fe, Co and Ni) can be used as the effective catalyst for the catalytic decomposition of hydrocarbons (such as methane, acetylene and alcohol) [as shown in reaction (5)] to produce CNTs. Therefore, the formation of Co@CNTs-G ternary hybrids should be the result of the reduction reaction between C_2H_2 and Co_3O_4/RGO. Moreover, according to the obtained results and previously reported models, the growth of CNTs over Co_3O_4/RGO should follow the tip growth mode and the formation mechanism of Co@CNTs-G ternary hybrids is given in Fig. 8. The possible pathways to grow Co@CNTs-G

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**Table 2. EM wave absorption properties of carbon-based and graphene-based hybrids reported in recent representative papers.**

| MAMs          | Optimum RL (dB) | Optimum thickness (mm) | Frequency range (GHz) (RL<−20 dB) | Reference |
|---------------|-----------------|------------------------|-----------------------------------|-----------|
| Co@C         | −52             | 3                      | 4.0–18.0                          | 56        |
| Ni/C          | −45             | 2                      | 4.0–14.2                          | 57        |
| (Fe, Ni)/C    | −26.9           | 2                      | 13.6–16.6                         | 58        |
| Fe/CNTs       | −25             | 1.2                    | 7.0–15.2                          | 59        |
| Fe/MWCNTs     | −39             | 4.27                   | 2.04–3.47                         | 60        |
| Co/MWCNTs     | −37             | 5.25                   | 2.35–3.51                         | 60        |
| Ni/MWCNTs     | −37             | 5.19                   | 1.83–3.07                         | 60        |
| G/CNTs        | −44.6           | 3                      | 6.0–9.0                           | 61        |
| Fe@G          | −45             | 3.0                    | 4.0–18.0                          | 33        |
| Co/G          | −47.5           | 2.0                    | 3.0–13.0                          | 62        |
| Ni/G          | −17.8           | 5.0                    | −                                | 63        |
| NiFe_2O_4@G   | −29.2           | 2.0                    | 5.7–17.5                          | 35        |
| MnO_2@Fe@G    | −17.5           | 1.5                    | −                                | 51        |
| ZnO/Fe@Fe_2O_3/G | −38.4         | 5.0                    | 5.9–15.2                          | 64        |
| G@Fe_2O_3@SiO_2@NiO | −51.5       | 1.8                    | 12.0–15.0                         | 65        |
| G400          | −65.6           | 2.19                   | 2.5–18.0                          | this work |
| G450          | −58.1           | 1.98                   | 5.0–15.8                          | this work |
| G500          | −41.1           | 2.03                   | 5.2–12.6                          | this work |
| G550          | −47.5           | 2.65                   | 2.6–11.0                          | this work |

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**Figure 8.** Schematic diagram for the formation mechanism of Co@CNTs-G ternary hybrids.
ternary hybrids are in the following manner: (1) the reduction reaction between Co$_3$O$_4$ and C$_2$H$_2$ to form Co nanoparticles; (2) the decomposition of C$_2$H$_2$ on the Co surface to generate carbon atoms; (3) the dissolution and diffusion of carbon atom around the catalyst; (4) carbon film grows along the surface of catalyst and results in the formation of Co@CNTs-G ternary hybrids.

In order to analyze the difference in obtained RL results and probable MA mechanism, the EM parameters, dielectric and magnetic loss abilities, attenuation constant and EM impedance matching are presented. Figure 9 gives the complex permittivity and complex permeability of Co/CNTs-G ternary hybrids in the 2.0–18 GHz frequency range. As shown in Fig. 9a and b, the obtained Co/CNTs-G hybrids exhibit the similar trends in the complex permittivity, which confirms further all the obtained samples are the same type of hybrids. Besides some fluctuations, the $\varepsilon'$ and $\varepsilon''$ values of Co/CNTs-G ternary hybrids are found to decrease with the frequency in the tested region. On the basis of the Debye theory, $\varepsilon'$ and $\varepsilon''$ can be described as:

$$
\varepsilon' = \varepsilon_\infty + \frac{\varepsilon_s - \varepsilon_\infty}{1 + \omega^2/\tau^2}
$$

$$
\varepsilon'' = \frac{\varepsilon_s - \varepsilon_\infty}{1 + \omega^2/\tau^2} + \frac{\sigma_{ac}}{\omega \varepsilon_0}
$$

where $\varepsilon_s$ is the static permittivity, $\varepsilon_\infty$ is the relative dielectric permittivity at the high frequency limit, $\omega$ is angular frequency, $\tau$ is polarization relaxation time, $\sigma_{ac}$ is the alternative conductivity and $\varepsilon_0$ is the dielectric constant in vacuum. According to the equations (6) and (7), one can find that the decreases of $\varepsilon'$ and $\varepsilon''$ are mainly attributed to the increase of $\omega$. As reported previously, the phenomenon can be considered as the polarization relaxation in the lower frequency range. Similar to the previously reported carbon-based composites, all the obtained hybrids exhibit the resonance peaks at ca. 5.8 and 13.8 GHz, which is favorable to the improvement of MA properties. Moreover, it can be seen that the complex permittivities of the obtained hybrids are as follows: G400< G550< G450< G500, which implies that the increasing CNT content in the obtained hybrid can enhance effectively the values of $\varepsilon'$ and $\varepsilon''$, and the relatively high CNT content in the obtained G550 will result in an evident reduction in these values. Similar to GO/CNTs-Fe$_3$O$_4$ composites, this can be explained that much more CNTs within limits in the obtained hybrids may increase the electric polarization and electric conductivity. Based on the obtained results, one can find that although the $M_T$ value of the obtained Co/CNTs-G ternary hybrids decreases with the increase of temperature (as shown in Fig. 6), the introduction of CNTs effectively improves the antioxidant capacity of Co nanoparticles and dielectric loss ability of as-prepared hybrids. Moreover, as we all know that the tube structure is more conducive to absorb EM wave, which is the major purpose of this study to introduce CNTs into the obtained hybrids. Therefore, compared to the previously reported results obtained from Fe@G, Co/G and Ni/G, the as-prepared Co/CNTs-G ternary hybrids exhibit enhanced MA capabilities. Figures 9c and d show the complex permeability of Co/CNTs-G ternary hybrids as a function of frequency. Similar to the complex permittivity, the obtained Co/CNTs-G ternary hybrids exhibit the similar trends in the frequency range.
complex permeability. And the $\mu''$ values of hybrids exhibit a peak at ca. 13.0 GHz, which can be ascribed to the natural resonance as reported elsewhere. Because of the little difference in magnetization, we can find that the discrepancy of complex permeability is unobvious.

Based on the Debye theory, if the second part of equation (7) is not taken into account, the relationship of $\varepsilon'$ and $\varepsilon''$ can be written as:

\[
\varepsilon' \varepsilon \varepsilon'' \varepsilon \varepsilon
\]

\[
\left( \varepsilon' - \frac{\varepsilon_1 + \varepsilon_\infty}{2} \right)^2 + (\varepsilon'')^2 = \left( \frac{\varepsilon_1 - \varepsilon_\infty}{2} \right)^2
\]

It corresponds to a circle center at $\left( \frac{\varepsilon_1 + \varepsilon_\infty}{2}, 0 \right)$, which is characteristic for Debye relaxation process. Therefore, the Debye relaxation process can be reflected in the plot of $\varepsilon'$ versus $\varepsilon''$, and each semicircle corresponds to one Debye relaxation process. As shown in Fig. 10, the Cole-Cole curves of the as-prepared hybrids indicate that each plot of $\varepsilon'$ versus $\varepsilon''$ contains many individual semicircles, which means the multirelaxations dielectric properties. According to the previous mechanisms, the multirelaxations are supported to originate from the multiple interfacial polarizations in the as-prepared Co@CNTs-G ternary hybrids.

Figure 11 shows a typical RL versus frequency for the as-prepared hybrids with the thickness of 2.0 and 3.0 mm. Besides the obtained G550, it can be seen that the minimum RL value of as-prepared hybrids moves towards the lower frequency region with the increasing temperature. Moreover, as mentioned aforementioned (as...
shown in Fig. 7), the minimum RL value moves towards the lower frequency region with the increasing thickness. Actually, these phenomena can be explained by the 1/2 wavelength equation:

\[ d_m = \frac{nc}{4f_m \sqrt{\mu_r |\varepsilon_r|}} \quad (n = 1, 3, 5 \ldots) \]  

Here, \( d_m \) and \( f_m \) are the matching thickness and frequency of RL peak. In terms of the same sample, one can find that \( d_m \) is inversely proportional to \( f_m \). Therefore, as shown in Fig. 7, the minimum RL value moves towards the lower frequency region with the increasing thickness. For the different samples, when \( d_m \) is kept constant, the complex permittivity and permeability are inversely proportional to \( f_m \). Based on the obtained results (as shown in Fig. 9), one can understand easily that the RL peak moves towards the lower frequency region with the increasing temperature besides the obtained G550. Generally, the obtained results indicate that the scope of absorption can be tuned by regulating CNT content in the obtained hybrids. Moreover, as reported previously\(^{76,77}\), graphene is more capable of forming conducive paths and the introduction of magnetic nanoparticles can lower the complex permeability of hybrids. Therefore, the addition of graphene and Co nanoparticles in hybrids is very important for the improvement and regulation of their MA properties. Compared the reported data with our obtained results\(^{60–62}\), one can find that the enhanced MA performance of as-prepared Co@CNTs-G ternary hybrids is attributed to the compensatory effect of Co nanoparticles, CNTs and graphene.

In order to analyze the possible enhanced MA mechanism, the dielectric and magnetic loss properties, attenuation constant and impedance matching of the obtained nanohybrids were investigated in details. As shown in Fig. 12a and b, one can find that all the obtained hybrids exhibit much larger values of \( \tan \delta_E \) than those of \( \tan \delta_M \), which implies that the EM attenuation is mainly due to dielectric loss. And the dielectric loss performance (as shown in Fig. 12a) of the hybrids presents the following tendency: G400 > G450 > G500 > G550. The difference in the values of dielectric loss should be related to the different CNT contents. Moreover, the obtained hybrids display excellent mutual compensation between dielectric loss and magnetic loss, and this effective compensation is very beneficial to enhance their MA capabilities\(^{60} \). According to equation (3), the \( \alpha \) values of hybrids were obtained and shown in Fig. 12c. It can be seen that all the as-prepared hybrids exhibit a similar variation curve and the difference in the \( \alpha \) values of G400, G450, G500 and G550 is very small. Actually, the \( \alpha \) value of as-prepared hybrid increases slightly with the rise of CNT content in the hybrid, and the \( \alpha \) value of G550 exhibits an evident fall when the CNT content in the obtained hybrid is high enough. In addition, compared to the previously reported MnO\(_2\)@Fe-G, Fe/MWCNTs, Co/MWCNTs and Ni/MWCNTs\(^{51,60}\), the \( \alpha \) values of the obtained ternary nanohybrids are much higher, and the high value of \( \alpha \) is conducive to improve EM wave absorption capability\(^{61} \). Based on the measured complex permittivity and permeability, the impedance matching ratios of the as-prepared ternary hybrids are obtained and the results are displayed in Fig. 12d. One can find that: (1) similar to the aforementioned results, all the as-prepared hybrids exhibit a similar variation curve; (2) the evident difference in the impedance matching ratios of G400, G450, G500 and G550 can be observed; (3) with the increase of temperature (from 400 to 500 °C), significant decrease can be observed in the impedance matching ratio; (4) the as-prepared G550 reverses the downward trend in the impedance matching ratio. Based on the aforementioned results, one can find that the enhanced MA capabilities of Co/CNTs-G ternary hybrids can be attributed to the good dielectric loss ability, excellent mutual compensation between dielectric loss and magnetic loss, high

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**Figure 12.** (a,b) Loss tangent, (c) attenuation loss, and (d) impedance matching of the obtained samples.
Co3O4/RGO. The CNT content in the as-prepared ternary hybrids could be regulated by controlling the temperature over 400, 450, 500 and 550 °C, respectively. For easy description, the samples obtained at 400, 450, 500 and 550 °C were denoted as G400, G450, G500 and G550, respectively.

**Table 3. RL values of the obtained nanohybrids at the calculated matching thickness (d_m) and the real thickness (d) in terms of geometrical effect.**

| Sample | d (mm) | RL (dB) | f_m (GHz) | d_m (mm) |
|--------|--------|---------|-----------|----------|
| G400   | 1.5    | −18.9   | 17.8      | 1.4      |
|        | 2.0    | −20.3   | 14.2      | 1.7      |
|        | 2.5    | −21.0   | 10.2      | 2.3      |
| G450   | 1.5    | −20.1   | 15.4      | 1.4      |
|        | 2.0    | −36.1   | 11.6      | 1.9      |
|        | 2.5    | −18.6   | 8.6       | 2.4      |
| G500   | 1.5    | −18.4   | 14.0      | 1.4      |
|        | 2.0    | −35.3   | 11.0      | 1.8      |
|        | 2.5    | −19.3   | 8.0       | 2.4      |
| G550   | 1.5    | −13.3   | 15.4      | 1.4      |
|        | 2.0    | −18.5   | 11.8      | 1.8      |
|        | 2.5    | −32.1   | 8.8       | 2.4      |

Methods

**Material preparation.** Co3O4/reduced graphene oxide (Co3O4/RGO) was purchased from XFNANO Materials Tech Co., Nanjing, China. The chemical reagents were analytically pure and used without further purification. Initially, 0.02 g of Co3O4/RGO was dispersed on a ceramic plate which was placed inside a quartz reaction tube and the tube furnace was heated from RT to the designed temperature (400, 450, 500 and 550 °C) in argon, respectively. Subsequently, acetylene was introduced into the reaction tube furnace at atmospheric pressure, and the pyrolysis of acetylene was carried out at a selected temperature for 2 h. Finally, after cooling to RT in argon, about 0.033, 0.055, 0.075 and 0.189 g of products (black in color) were collected on the plate at 400, 450, 500 and 550 °C, respectively. For easy description, the samples obtained at 400, 450, 500 and 550 °C were denoted as G400, G450, G500 and G550, respectively.

**Measurement.** The samples were examined on an X-ray powder diffractometer (XRD) at RT for phase identification using CuKα radiation (model D/Max-RA, Rigaku). Raman spectroscopic investigation was performed using a Jobin-Yvon Labram HR800 instrument with 514.5 nm Ar⁺ laser excitation. The morphology investigations were examined using a transmission electron microscope (TEM) (model Tecnai-G20, operated at an...
accelerating voltage of 20 kV), and field emission scanning electron microscope (FE-SEM) (model ZEISS SUPRA-40, JSM-7500F and S-3400, operated at accelerating voltages of 5 kV). Fourier transform infrared (FT-IR) spectroscopy of samples (in KBr pellets) was recorded using a Nicolet 510 F spectrometer. Atomic force microscopy (AFM) measurements were carried out by using Nanonics MV4000. The magnetic properties of samples were measured at 300 K using a Quantum Design MPMS SQUID magnetometer (Quantum Design MPMS-XL) equipped with a superconducting magnet capable of producing fields of up to 50 kOe. For microwave measurement, 30% of the as-prepared sample was mixed with paraffin and pressed into coaxial clapper in a dimension of outer diameter of 7.0 mm, inner diameter of 3.0 mm, respectively. The complex permittivity ($\varepsilon' = \varepsilon'' - j\varepsilon''''$) and complex permeability ($\mu' = \mu'' - j\mu''''$) of the composite were measured in frequency range of 2–18 GHz over an Agilent E8363B vector network analyzer.

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Author Contributions
Q. Hu and X.S. Qi are the joint first authors, they contributed equally to this work. Q. Hu, H.B. Cai and X.S. Qi designed and carried out the experiments. R. Xie, Z.C. Bai and Y. Jiang collected the experimental data. Results were analyzed and interpreted by Q. Hu, X.S. Qi, S.J. Qin, W. Zhong and Y. W. Du. The manuscript was written by Q. Hu, X.S. Qi and W. Zhong.

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