Supporting Materials and Methods

Photocrosslinking to DNA substrates

The ASV IN derivatives were modified by coupling with two photoactivatable thiol-specific compounds, a carbene-generating compound, N-bromoacetyl-N'-[2,3-dihydroxy-3-[3-((trifluoromethyl)diazirin-3-yl)phenyl]propionyl]ethylenediamine (BATDHP) [1] or a nitrene-generating compound, azidophenacylthiopyridine (APTP) [2]. Carbenes are among the most reactive moieties known. They are capable of insertion into any chemical bond present in a biomolecule, including aliphatic chains and aromatic rings. Having nanosecond half-lifetimes, carbenes rapidly form covalent bonds with neighboring atoms. Although electrophilic, carbenes are so highly reactive that in the absence of strong nucleophiles they attack even the C-H bonds (over 80% crosslinking to cyclohexane) [3,4]. The high level of reactivity of carbenes with buffer components usually precludes high yields of crosslinked products but their low specificity makes them ideal for identification and localization of members and parts of affinity complex. Nitrenes such as those generated from the azide reagent, APTP, are less reactive and tend to undergo intramolecular rearrangements that lead to even less reactive products. They crosslink primarily to nucleophiles such as amino groups and, in a non-nucleophilic environment, can remain active for periods of up to several minutes. This makes them less reliable for detection of close interactions since selective crosslinking may preferentially occur to a relatively distant nucleophilic group that is only occasionally in the vicinity of the crosslinker. The efficiency of crosslinking with nitrenes is thus higher, but there is a possibility of bias towards interactions with nucleophilic groups that are not relevant to the complex under study [4].

Oligonucleotide synthesis with 8-amino-3,4-dithiaoctyl tether at N² of dG

The oligonucleotides were prepared on commercial dC-cpg (40-50 μmol/g) using a standard synthesizer protocol for generation of the 21-base sequence at the 3’-end. The 5’-DMT protected phosphoramidite [5] derived from O⁶-(2-p-nitrophenylethyl)-2-fluoro-2'-deoxyinosine was coupled manually to the 5’-end of the oligonucleotide bound to the support [6]. A typical 2-μmole synthesis utilized 20 mg (21 μmol) of the phosphoramidite and 150 μL of 0.5 M 4,5-dicyanoimidazole in acetonitrile for 16 h at rt.; yield was estimated from the recovery of DMT cation after deprotection. End capping with acetic anhydride was omitted after the manual coupling step [6], and the support-bound oligonucleotide was directly oxidized (1 M tert-butyl hydroperoxide in dichloromethane, 30 s) and returned to the synthesizer for addition of the remaining 8 residues by the standard automated synthesis procedure. After removal of the 5’-DMT protecting group, the linker was coupled to the support-bound oligonucleotide using a
modification of the procedure of Erlanson et al. [5]. The support-bound oligonucleotide was treated with 45 mg (178 μmol) of 3,3'-dithiobis(propylamine) dihydrochloride [7], which was prepared as described for the 4-carbon homolog [8] 60 μL (600 μmol) triethylamine and 100 μL H₂O for 16 h at room temperature. In the course of preparing O⁶-(2-p-nitrophenylethyl)-2-fluoro-2'-deoxyinosine, we observed cleavage of the nitrophenylethyl protecting group in the presence of wet tert-buty ammonium fluoride. After reaction with dithiobis(propylamine), addition of concentrated NH₄OH (1.5 mL) containing 20 μmol tert-buty ammonium fluoride to the beads and solution, followed by heating at 60 °C for 3 days, resulted in complete debllocking of the oligonucleotide; this procedure avoided the DBU/formamide cleavage step and accompanying formylation [5] of the free amino group of the tether. After filtration, the oligonucleotide solution was dialyzed against 0.1 M triethylammonium acetate buffer (pH 6.0) overnight to remove excess amines. The oligonucleotide was purified by HPLC on a Hamilton PRP-1 column (7 μm, 10 x 250 mm) eluted at 3 mL/min with a linear gradient of acetonitrile in 0.1 M (NH₄)₂HCO₃ buffer (pH 7.5) that increased the acetonitrile concentration from 0 to 17.5% over 20 min; tᵣ 16.3 min.

**Synthesis of the 3’ modified viral end substrates for crosslinking to the active site IN derivatives.**

Oligonucleotides 5’-GAGTATTGCATAAGACTAC-A*-3’ where A* represents the following structures:

![Structure 1](image1)

![Structure 2](image2)

were synthesized following a standard protocol using modified solid support of the following structure: CPG-700-LCAA-NH-CO-(CH₂)₂-COO-(CH₂)₂-SS-(CH₂)₃-O-DMTr

For debllocking a standard mixture of ammonia and methylamine also contained 50mM DTT for disulfide cleavage.
Schematics of nucleotide synthesis:

3’-end N-mercaptoethyl morpholine-adenosine was synthesized according to Stirchak et al. [9]. To complete the morpholine ring we used cystamine acetate, (NH$_2$-(CH$_2$)$_2$-SS-(CH$_2$)$_2$-NH$_2$)Ac$_2$, for amino component. The disulfide was cleaved by DTT to yield corresponding mercaptane:

\[
P \rightarrow O-(CH$_2$)$_2$-SS-(CH$_2$)$_3$-O-DMTr
\]

mercaptoethanol

\[
P \rightarrow O-(CH$_2$)$_2$-SH + HS-(CH$_2$)$_3$-O-DMTr
\]
dipyridyl disulfide

\[
P \rightarrow O-(CH$_2$)$_2$-S-S-Py
\]
activated nucleoside

that was then attached to modified solid support. The support modification consisted of mercaptoethanol treatment to achieve corresponding polymer-mercaptane, followed by activation with dipyridyl disulfide.

Isolation of the synthesized oligonucleotide was performed as described above. Both oligonucleotides were purified by RP-HPLC and their structures confirmed by mass-spectrometry (MALDI-TOF).
References

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