Bioactive Titanium-Hydroxyapatite Composites by Powder Metallurgy Route

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Abstract: Titanium (Ti) and its alloys have become the most promising biomaterials due to their low elastic modulus, high corrosion resistance, and relatively long-lasting ability in a physiological environment. Bioactive implants enhance the tissue interactions at the surface of the implants and promote a higher healing rate. However, titanium exhibits bio-inert nature. Hence in the present study, hydroxyapatite (HA), a well-known bioceramic, has been selected to disperse into Ti with an aim to develop bioactive Ti-based implants. Ti-HA composites with 5% and 10% HA were successfully produced by high-energy ball milling for 20 h followed by sintering (at 850 °C). Fine-grained composites were successfully produced and were found to be free from any impurities. The composites were immersed in simulated body fluid (SBF) for 4 weeks to investigate the in vitro bioactivity. From the XRD studies and scanning electron microscope observations, the presence of HA in the composite enhanced the bioactivity as reflected with higher Ca/P mineral phases on the surface of the composites compared with pure Ti substrate. From the results, it can be concluded that the bioactive nature of Ti can be enhanced by reinforcing HA to manufacture medical implants with a higher healing rate.

Keywords: titanium; hydroxyapatite; bioactivity; composite; biomineralization; wettability.

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1. Introduction

Titanium (Ti) and its alloys are the important materials to manufacture orthopedic implants owing to their promising properties compared with steels and Co alloys [1]. Titanium exhibits significantly lower young's modulus than steel implants and helps reduce stress shielding phenomena, which is usually associated with the metallic implants that possess higher young's modulus than bone [2-4]. Ti also reduces the difficulty in diagnosing patients using MRI scanning as Ti is non-magnetic compared with steel implants. The corrosion resistance of Ti is relatively higher in the physiological solutions and gives longer life to the implant. However, Ti suffers from insufficient bio-activity, which can strongly bond with the local hard tissue [5, 6]. From the context of load-bearing requirements, Ti and its alloys possess sufficient mechanical properties, and no further improvement is sought to increase mechanical strength. Whereas in the context of tissue interactions with the implant surface, there exists a need to develop tailored surfaces to promote a higher healing rate. Several strategies, such as providing surface coatings, surface chemical treatments, microstructure modifications, etc., have been adopted to improve the bioactivity of Ti [7-12]. Furthermore, the surface of titanium implants was tailored by providing special coatings to exhibit anti-infection properties, antibacterial properties, better cell response, and improved bioactivity [13-16]. It is also reported that the
porosity, surface morphology, texture, and surface energy significantly influence the cell adhesion mechanisms at the surface of biomedical implants [17-19]. On the other hand, altering the composition of the implant by adding appropriate reinforcing particles into Ti is a promising strategy that enhances the bioactivity of the implant. This route also eliminates the issues associated with the bonding quality between the coating layer and the Ti substrate. A few earlier reports demonstrate using hydroxyapatite as a coating material or dispersing phase to produce bioactive composites [20-23]. These studies demonstrate developing Ti-based functionally graded materials through powder metallurgy. High energy ball milling is a powder processing technique that helps to reduce the particle size and to mix the powders uniformly [24]. In addition to uniformly mixing the powders, mechanical alloying of powders can also be done by high-energy ball milling. Hydroxyapatite (HA) is a ceramic phase of calcium and phosphorous usually found in natural bone in the form of nanocrystals [25]. Therefore, adding HA to Ti helps to promote better cell activities at the implant surface, which further leads to a higher healing rate. Earlier reports demonstrate the positive effect of the addition of HA to different implant materials on improving cell adhesion and proliferation [26-28]. Hence, in the present work, lab-prepared nano-HA powder was selected to disperse Ti by ball milling followed by sintering. Then the role of added HA on enhancing the wettability and bioactivity was assessed by in vitro bioactivity studies conducted using simulated body fluid.

2. Materials and Methods

Pure Ti powder (Merc, India) with an average particles size of 114 ± 6.2 µm was used as the starting material. Lab-prepared nano-HA was used as reinforcement material in the present work. The composition of 5% and 10% HA (by wt.) remaining pure Ti was prepared, and ball milled for 20 h. Ball milling was carried out using a tungsten carbide vial of 80 ml volume and tungsten carbide balls of 10 mm dia. The powder to ball weight ratio was maintained as 20:1. Ball milling was done in ethanol medium. For every 1 h of ball milling, 30 min idle time was given to cool the equipment. The ball-milled powders were consolidated in a hydraulic press by applying a 200 MPa load with a pellet size of 10 mm dia. Then the green composites were sealed in vacuum tubes and sintered in a box furnace at 850 °C. Figure 1 schematically explains the sequence of steps involved in developing Ti-HA composite in the present work.

Commercially available pure titanium (CP-Ti, grade 2) was also used to compare with the produced composites (Ti-5HA and Ti-10HA). All the samples were polished as per the standard metallographic protocol and etched with Kroll's reagent. Microstructural observations were done by using an optical microscope and electron microscope (SEM, TESCON, Czech Republic) attached with an energy-dispersive X-ray (EDS) facility. Nano-HA crystals were characterized by transmission electron microscope (TEM, FEI, USA). All the samples were subjected to X-ray diffraction analysis (D8, Bruker, USA). In order to assess the surface wettability of the samples, water contact angle measurements were done on the surface of all the samples by using distilled water (2 ml) as the solvent. Then the bioactivity of the samples was investigated by immersion studies carried out in simulated body fluid (SBF). The procedure and the ion concentration of SBF can be referred to elsewhere [29, 30]. Each sample was immersed in a solution of 50 ml, and the containers were placed in a water bath that was maintained at 37°C for 4 weeks. Then the samples were collected and gently rinsed with distilled water and dried in the open air. The surface of the immersed samples was subjected to
SEM and XRD studies to analyze the mineral phases which were deposited on the surface from the SBF.

![Schematic illustration of developing Ti-HA composites by ball milling and sintering.](https://doi.org/10.33263/BRIAC124.53755383)

Figure 1. Schematic illustration of developing Ti-HA composites by ball milling and sintering.

3. Results and Discussion

The optical microscope image of CP-Ti is shown in Figure 2 (a). From the microstructure, the average grain size was measured as 43 ± 4.7 µm. Figure 2(b) shows the SEM image of pure Ti powder used in the present study. The corresponding EDS analysis (Figure 2 (c)) confirms the elemental composition in the procured Ti powder. The Ti particles were observed with irregular shapes. From the TEM image of nano-HA, the crystals were observed with acicular morphology with significant length (67 nm) to thickness (25 nm) variation. The corresponding SAED pattern shows the appearance of spots as ring structures, which is a typical pattern for nanocrystals. These observations confirm the nanocrystalline nature of the HA used in the present work to develop Ti—HA composites.

![Figure 2](https://biointerfaceresearch.com/)

Figure 2.(a) microstructure of pure Ti, (b) SEM image of Ti powder, (c) corresponding EDS analysis, (d) TEM bright field image of nano-HA, and (e) corresponding selective area electron diffraction (SAED) pattern.
From the XRD analysis (Figure 3), the crystallite size of nano-HA was calculated as 48 µm, which also confirms the nano-level of the HA used in the present work and supports the TEM observations of HA. Typical XRD of Ti-10HA shows all the peaks corresponding to Ti and peaks corresponding to HA. Additionally, no new peaks were identified apart from the peaks corresponding to Ti and HA, confirming that the HA phase added to Ti was stable after sintering. The stability of HA is crucial to retain the benefits of adding HA to Ti in developing these composites.

![Figure 3. XRD analysis of the samples.](image)

Usually, HA is stable up to 1000 °C without any phase transformation. Since the sintering temperature in the present work was 850 °C, no degradation of HA to any other unusual phase is expected. Figure 4 shows the microstructure of sintered Ti-HA composites. The grain size was measured as 2.3 ± 0.7 µm and 1.9 ± 0.6 µm for Ti-5HA and Ti-10HA, respectively. A few grains with sub-micrometer levels were also observed. The grain refinement observed in the composites can be attributed to the size reduction of Ti particles during the ball milling. It is reported that ball milling for longer periods significantly decreases the particles’ size and results in nanocrystals [24]. As observed in the present work, decreased particle size after high-energy ball milling helped achieve a smaller grain size after sintering.

![Figure 4. SEM observations: (a) Ti-HA and (b) Ti-10 HA composite.](image)

The contact angle measurements revealed the increased wettability for the composites compared with CP-Ti. Figure 5 shows the typical photographs of water droplets on the surface...
of the samples. Contact angle (θ) was measured for the samples and listed in Table 1. All the samples exhibited hydrophilic nature as reflected from the lower contact angles (θ). It is desired to have more hydrophilicity for the surfaces of implants to promote wettability. Surfaces with high wettability enhance biomineralization and cell adhesion and promote a high healing rate [31, 32]. Lower contact angles were measured for the composites compared with CP-Ti, indicating higher wettability for the composites than CP-Ti. The presence of bioactive nano-phases at the surface also promotes a higher rate of cell regeneration and growth [31].

![Figure 5](https://doi.org/10.33263/BRIAC124.53755383)

**Figure 5.** Photographs of water droplets on the samples: (a) CP-Ti, (b) Ti-5HA, and (c) Ti-10HA.

| S.No. | Sample   | Contact angle (θ) |
|-------|----------|-------------------|
| 1     | CP-Ti    | 64.5 ± 3.2        |
| 2     | Ti-5HA   | 59.1 ± 2.8        |
| 3     | Ti-10HA  | 57.2 ± 1.7        |

The surface morphologies and corresponding EDS analysis of the samples after 4 weeks of immersion in SBF are shown in Figure 6. Deposition of mineral phases on the surface of the immersed samples can be seen, and the corresponding EDS analysis confirms the presence of Ca and P elements in addition to Ti. The presence of Ca and P suggests the formation of calcium-based mineral phases on the surface of the SBF. XRD analysis (Figure 7) also confirms the deposited phases as hydroxyapatite. Higher wettability and presence of nano-HA in the composite accelerated mineral deposition from the SBF. Usually, smaller grain size is associated with a larger fraction of grain boundary. Grain boundaries are high-energy regions, and therefore, fine grain structure gives higher surface energies. Therefore, the nucleation of apatite crystals from the SBF is accelerated at the high-energy sites. Furthermore, nano-HA in the composites acts as nuclei and contributes to the growth of apatite from the SBF. This is similar to the reported work developing bioactive composites by using nano-HA as dispersing phase [33]. Furthermore, nano-HA's presence was found to be influential in enhancing the tissue response for quick healing due to the osseointegration exhibited by nano-HA. Therefore, smaller grain size, increased wettability, and presence of nano-HA in the composites improved the Ca/P mineral phase deposition, indicating increased bioactivity. Figure 8 shows the deposition of the Ca/P mineral phase as a thick layer on a selective region of Ti-10HA. These findings suggest that the Ti can be made bioactive by incorporating a sufficient amount of HA.

The present study successfully demonstrates producing Ti-HA composite by powder metallurgy route. The need for bioactive titanium to manufacture bone fixing plates, screws, and dental implants is growing to promote a higher healing rate. Bioactivity is a phenomenon connected with the surface of the implant and the tissue. Hence, providing surface coatings may improve the bioactivity as the tissue interactions with the implant are initiated from the surface. However, the success of the coating depends on the quality of the coating and the bonding level between the implant substrate and the coating material.
Figure 6. SEM observations of the surface of the samples after immersion (4 weeks): (a) CP-Ti, (b) corresponding EDS analysis, (c) Ti-5HA, (d) corresponding EDS, (e) Ti-10HA and corresponding EDS analysis.

Figure 7. XRD patterns of immersed samples (4 weeks).
Particularly for implants where temperature variations are common, the coated material (ceramic) and the substrate (metal) experience different levels of thermal expansions due to the mismatch in thermal properties, and the bonding between the coating material and the implant is affected and may lead to failure of the coating. For example, in dental implants, where consuming cold and hot beverages gives a considerable change in the temperature, the coating on the implant may fail due to the aforementioned reason. Therefore, producing the Ti implant with inherent bioactivity property by incorporating bioactive imparting phases into the implant can be a viable strategy. As demonstrated in the present work, adding nano-HA as a reinforcing phase to manufacture Ti-HA composite helps produce implants with enhanced bioactivity. As the content of HA is increased to 10%, the effect was observed as higher. Further investigations on finding the role of added nano-HA on mechanical properties are needed to be studied to manufacture bioactive Ti-HA implants.

4. Conclusions

The present work incorporated nano-hydroxyapatite (HA) powder into Ti by high-energy ball milling and sintering to develop bioactive Ti-based implants. No significant contamination and impurity was observed in the composites from the XRD analysis. Compared with the starting particle size, the average grain size of the sintered composite was significantly decreased. Contact angle measurements confirmed the increased wettability for the composites compared with reference CP-Ti. From the bioactivity studies conducted in SBF for 24 days, a higher rate of mineral deposition was observed on the surface of Ti-HA composites, and the effect was more on the Ti-10 HA composite. Smaller grain size increased wettability, and the addition of nano-HA were the prominent reasons behind the enhanced bioactivity, as reflected from the deposition of more Ca/P mineral phase from the SBF. Hence, it can be concluded that by dispersing HA, bioactive Ti-based implants can be successfully produced through the powder metallurgy route to manufacture implants for bone and dental applications.

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Conflicts of Interest

The authors declare no conflict of interest.

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