Abstract

Serpentinization of ultramafic rocks in the sea and on land leads to the generation of alkaline fluids rich in molecular hydrogen ($H_2$) and methane (CH$_4$) that favour the formation of carbonate mineralization, such as veins in the sub-seafloor, seafloor carbonate chimneys and terrestrial hyperalkaline spring deposits. Examples of this type of seawater–rock interaction and the formation of serpentinization-derived carbonates in a shallow-marine environment are scarce, and almost entirely lacking in the geological record. Here we present evidence for serpentinization-induced fluid seepage in shallow-marine sedimentary rocks from the Upper Cretaceous (upper Campanian to lower Maastrichtian) Qahlah Formation at Jebel Huwayyah, United Arab Emirates. The research object is a metre-scale structure (the Jebel Huwayyah Mound) formed of calcite-cemented sand grains, which formed a positive seafloor feature. The Jebel Huwayyah Mound contains numerous vertically orientated fluid conduits containing two main phases of calcite cement. We use C and O stable isotopes and elemental composition to reconstruct the fluids from which these cements precipitated and infer that the fluids consisted of variable mixtures of seawater and fluids derived from serpentinization of the underlying Semail Ophiolite. Based on their negative δ$^{13}$C values, hardgrounds in the same section as the Jebel Huwayyah Mound may also have had a similar origin. The Jebel Huwayyah Mound shows that serpentinization of the Semail Ophiolite by seawater occurred very soon after subduction and marine transgression, a process that continued through to the Miocene, and, with interaction of meteoric water, up to the present day.

1. Introduction

Serpentinization is an exothermic geochemical reaction that takes place in marine and terrestrial settings where ultramafic rocks interact with seawater or meteoric waters, causing the hydration of the primary mafic mineralogy (e.g. olivine, pyroxenes) to serpentine, brucite and magnetite (e.g. Sleep et al. 2004; Bach et al. 2006). This process generates large quantities of hydrogen ($H_2$), which in turn results in the formation of abiotic methane (CH$_4$) through the reduction of carbon dioxide (CO$_2$) or bicarbonate (HCO$_3^-$) (e.g. Proskurowski et al. 2008; Etiope et al. 2011). Fluids deriving from the reaction are enriched in hydrogen and methane, and are usually hyperalkaline (pH = 11) (Kelley et al. 2001; Palandrini & Reed, 2004). They can also contain high concentrations of Ca from the breakdown of Ca-rich pyroxene minerals during serpentinization (Bruni et al. 2002; Chavagnac et al. 2013). When serpentinization-derived fluids come close to the seafloor or land surface, they mix with seawater or meteoric waters and carbonate minerals commonly precipitate, which are usually calcite and/or aragonite, often together with brucite (Ludwig et al. 2006). These authigenic carbonates can have a variety of morphologies, from veins through to highly porous chimney-like structures many metres tall (Früh-Green et al. 2003; Ludwig et al. 2006). Serpentinization-derived fluid emission has now been recognized in multiple geographic areas and tectonic settings where ultramafic rocks are exposed on land (as ophiolites) or in the sea. These include in the deep-sea, oceanic core complexes (e.g. Lost City on the Atlantic Massif; Kelley et al. 2001; Früh-Green et al. 2003), serpentinite mud volcanoes (e.g. Mariana fore-arc seamounts; Fryer et al. 1985) and rifted continental margins in the initial stages of ocean basin development (e.g. Iberian Margin; Agrinier et al. 1996; Klein et al. 2015). Examples also occur where ophiolites are in shallow-marine settings (e.g. Bay of
Prony, New Caledonia and offshore Elba, Italy; Monnin et al. 2014; Meister et al. 2018), and are best known from terrestrial ophiolites (e.g. Semail Ophiolite, Oman, Del Puerto Ophiolite, California and Zambales Ophiolites, Philippines; Neil & Stanger, 1984; Abrajano et al. 1988; Blank et al. 2009; Chavagnac et al. 2013). Whilst serpentinization-derived fluid seepage and associated carbonate mineral formation is now well known in modern settings, direct evidence for these processes in the geological record is sparse and comes largely from carbonate veins in ophiolites, often called opalcalcites (e.g. Lavoie & Chi, 2010; Klein et al. 2015; Lafay et al. 2017; de Obeso & Kelemen, 2018, 2020; Cooperdock et al. 2020). Nevertheless, the process is of great interest because of the astrobiological implications of serpentinization through the abiogenic formation of organic molecules (Proskurovski et al. 2008; Lang et al. 2010).

Here we describe an ancient inferred example of shallow-marine serpentinization-related seepage in the Upper Cretaceous (upper Campanian to lower Maastrichtian) Qahlah Formation of the border region between Oman and the United Arab Emirates. We suggest that the occurrence, distribution and composition of its obduction (Glennie 1988; Blank et al. 2009; Smith et al. 1995) and marls rich in Loftusia (Skelton et al. 1990; Smith et al. 1995; Abdelghany, 2006; Abbasi et al. 2014). The age of the Semail Ophiolite is considered to be approximately coeval with the time of its obduction (Glennie et al. 1973), and relatively little time elapsed between the final stages of this process (Searle & Cox, 1999) and the deposition of the Qahlah Formation. The formation is time transgressive and comprises coarse clastic shallow-water sediments, variously fluvialite to marine, which are difficult to date precisely, but are usually considered to be latest Campanian or early Maastrichtian in age based on the presence of rudist bivalves, corals and especially species of large benthic foraminifera such as *Loftusia* (Skelton et al. 1990; Smith et al. 1995; Abdelghany, 2006; Abbasi et al. 2014).

At Jebel Huwayyah, 10 km northeast of the town of Al Ain in Abu Dhabi (Fig. 1a), up to 24 m of the Qahlah Formation crops out (Fig. 1b, c). The base of the formation and its presumed contact with the underlying Semail Ophiolite are not exposed here, but at other localities to the north, Qahlah Formation sediments lie directly on top of weathered ophiolitic rocks. The Qahlah Formation at Jebel Huwayyah consists predominantly of coarse clastic sediments (conglomerates and sandstones), sometimes cross-bedded. Unlike at most other localities, the Qahlah Formation sediments at Jebel Huwayyah have an appreciable carbonate content, including thin fringing calcite cement layers on some chert cobbles, laterally discontinuous hardgrounds formed by synsedimentary lithification in the lower and middle part of the section (Beds 2, 7 and 9 of Smith et al. 1995) and marls rich in *Loftusia* species and other large benthic foraminifera in the upper part of the section (Smith et al. 1995; Alsharan & Nasir, 1996; Wilson & Taylor, 2001; Abdelghany, 2006). Abdelghany (2006) interpreted the part of the Jebel Huwayyah section below the *Loftusia* beds as being deposited in fluviatile to beach environments, but this seems to contradict the finding by Smith et al. (1995) of rare shell lenses of the marine oyster *Acrostrea* in the basal Beds 1 and 3, and rudist fragments and pebbles encrusted with *Acrostrea*, bryozoans and corals in Bed 7, suggesting rather that the entire lower part of the section has a shallow-marine origin. Within Bed 2 of the Jebel Huwayyah section is a mound-shaped structure with distinctive carbonate cement-filled vugs and tubular structures (Fig. 1d, e) that is the subject of this study, which we here call the Jebel Huwayyah Mound (JHM).

### 3. Methods

For petrographic studies, four polished thin-sections were made of the JHM matrix sediments and tubular structures, and two Qahlah Formation hardground specimens from Bed 2. These were viewed and photographed with light microscopes at Leeds University and MARUM. Subsequently, selected polished thin-sections were stained with Feigl’s solution, Alizarin Red-S and potassium ferrocyanide for identification of carbonate minerals. In addition, one unstained polished thin-section from the JHM was carbon coated and viewed with a Cameca SX-50 microprobe at Leeds University. Another thin-section was examined using a Zeiss Supra30 field emission gun scanning electron microscope (FEG-SEM) at the University of Bremen.

A number of distinctive carbonate cement phases were identified during the petrographic studies of the JHM tubular structures and, together with the matrix sediments and the hardgrounds, these were micro-drilled for C and O stable isotope analyses. CO₂ for isotopic analysis was quantitatively released from carbonate samples by the standard procedure of overnight reaction in a vacuum with 100 % phosphoric acid at 100 ºC. Gases were then analysed on a VG SIRA 10 mass spectrometer at the Scottish Universities Environmental Research Centre, monitoring mass: charge ratios 44, 45 and 46. Analytical raw data were corrected using standard procedures (Craig, 1957). All isotope data are reported in the standard δ-notation in ‰ relative to Vienna Pee Dee Belemnite (V-PDB). The error of reproducibility, based on complete analysis of internal standards (including acid digestion) was ±0.1 ‰ for δ¹³C and ±0.2 ‰ for δ¹⁸O values. Formation temperatures were calculated after Kim & O’Neil (1997).

Minor (Mg, Si, K, Mn, Fe and Sr) and trace (Cr, Ni, Y, La, Ce, Nd, Sm, Eu, Gd, Dy, Er, Yb and Pb) elemental compositions of carbonate minerals were analysed from four polished thin-sections of the JHM with a NewWave UP193 solid state laser ablation system (λ = 193 nm) coupled to a ThermoFinnigan Element 2 sector field inductively coupled plasma mass spectrometer (ICP-MS) at the Department of Geosciences, University of Bremen. In order to avoid surface contamination, each sample was pre-ablated five times with a spot size of 120 μm. Carbonate samples were ablated by a laser beam (irradiance of ~0.14 GW/cm²) with a pulse rate of 5 Hz and a spot size of 100 μm. Data were calibrated against the NIST SRM 612 glass standard reference (Pearce et al. 1997) using ⁴⁴Ca as the internal standard and assuming a Ca concentration of 40.04 wt % for calcite. Multiple laser analyses of individual calcite samples revealed a standard reproducibility of <5.5 %.

For reconstructing the past fluid Mg/Ca and Sr/Ca ratios from the JHM carbonate-filled tubes, we used the same approach as Rausch et al. (2013). In brief, we used an average formation temperature for the various carbonate cement phases using the
equation of Friedmann & O’Nell (1977), and the $\delta^{18}O$ value of past seawater of $-1\%$ for samples older than 15 million years (Muehlenbuchs, 1998). The distribution coefficients for Mg/Ca were calculated after Rimstidt et al. (1995), whereas those for Sr/Ca were calculated as in Rausch et al. (2013).

4. Results

4.a. Sedimentology and petrography

The JHM is a partially eroded, roughly spherical structure ~1 m in diameter and is formed of medium-grained lithic sandstone cemented by inclusion-rich calcite (Fig. 1d, e). The mound originates from an underlying bed of sandstone, forming a positive structure draped by overlying conglomerates (Fig. 1d). The sand grains forming the JHM are sub-angular to sub-rounded in shape and are a mixture of lithologies (Figs 2, 3), including in order of decreasing abundance: haematite-stained serpentine, quartz, opaque chert (sometimes with radiolarian fossils; Fig. 2b, c), magnetite, chromium spinel, biotite and micritized biogenic carbonate grains (Fig. 2d). Numerous hollow vugs and tubular structures occur within the JHM, but are not found in the underlying or host sandstone, or the overlying conglomerates (Fig. 1e). The tubes are roughly cylindrical, 0.5 to 7 mm in diameter (Fig. 3), and have an orientation that is roughly perpendicular to the base of the mound (Fig. 1e). The tubes have a visible length of up to 80 mm, but because of the eroded nature of the JHM some could have originally been longer. One tube cut longitudinally has a flat base containing a small number of significantly larger fresh serpentine grains (up to 2 mm in length), and below this base, a number of converging, thin calcite-filled channels (Fig. 2a). This tube also shows a distinctive orientation of the surrounding sand grains, which over the space of a few millimetres become increasingly rotated towards the tube, so that some grains come to lie parallel with the tube wall (Fig. 2a). The vugs and tubes in the JHM are lined by two main phases of calcite cement: (1) an early inclusion-rich, fibrous calcite that forms isopachous rims 0.2–0.3 mm in thickness lining tube walls, which we hereafter refer to as early cement (Figs 2a, 3) and (2) a later phase of inclusion-free equant, blocky calcite, which we hereafter refer to as late cement (Figs 2a, 3). The early cement we interpret to be the same calcite generation that cements the grains forming the mound (Fig. 2b–d) on the basis of appearance, continuity from grain-cementing to isopachous rims (Fig. 2a) and stable isotope values (Table 1). In some tubes the late cement reveals patches of microsparitic calcite (Fig. 2a), and in some of the larger diameter tubes and vugs the late cement does not entirely fill their interiors, leaving hollow spaces (Figs 1e, 3a).

4.b. Carbon ($\delta^{13}C$) and oxygen ($\delta^{18}O$) stable isotopes

The $\delta^{13}C$ values of the early cement in the JHM tubes range from $-4.2\%$ to $-3.3\%$ ($n = 4$), whereas the $\delta^{13}C$ values of the late cement range from $-7.9\%$ to $-5.3\%$ ($n = 9$; Table 1; Fig. 4). A single $\delta^{13}C$ value from the microsparitic cement ($-5.8\%$) is similar to the range of the late cement with which it is associated. The $\delta^{13}C$ values of the calcite cementing the grains forming the mound are between $-3.5\%$ and $-3.3\%$ ($n = 3$). The $\delta^{18}O$ values fall between $-3.1\%$ and $-1.3\%$ for the early calcite cement, between $-3.1\%$ and $-1.1\%$ for the late cement, and between $-3.0\%$ and $-2.2\%$.
for calcite cementing the mound; one sample of microsparitic cement yielded a $\delta^{18}O$ value of $-1.5$‰. The $\delta^{13}C$ values of the Qahlah Formation hardgrounds range from $-9.1$‰ to $-4.8$‰, and corresponding $\delta^{18}O$ values are between $-4.1$‰ and $-0.3$‰. The oxygen isotopic compositions were used to estimate the precipitation temperature of calcite (Table 1), using the empirical relation between $\delta^{18}O_{\text{calcite}} - \delta^{18}O_{\text{water}}$ and temperature by Kim & O’Neil (1997).

4.c. Mg/Ca and Sr/Ca ratios

The elemental composition of the calcite-filled tubular structures in the JHM reveal distinct Mg/Ca and Sr/Ca ratios for the two main cement phases (Fig. 5; online Supplementary Material Table S1).

The early cement has Mg/Ca ratios of between 9.53 and 19.44 mmol/mol and Sr/Ca ratios of 0.25 to 0.47 mmol/mol, whilst the Mg/Ca ratios for the late cement range from 0.85 to 6.45 mmol/mol and the Sr/Ca ratios are between 0.04 and 0.17 mmol/mol. The Mg/Ca and Sr/Ca ratios for the microsparitic cement are, with one exception, intermediate between the values of the early and late cements. The reconstructed Mg/Ca and Sr/Ca ratios of the parental fluid from which the calcites precipitated are listed in online Supplementary Material Table S1.

4.d. Rare earth elements and yttrium

Rare earth element (REE) and yttrium (Y) concentrations of the calcite cements (Fig. 6; online Supplementary Material Table S2)
show an overall flat post-Archaean Australian shale (PAAS)-nor-
malized pattern for the late phase with small positive Y anomalies. The early cement shows slight light REE depleted patterns and
lacks a positive Y anomaly. Both of the main cement phases show
weak negative Ce anomalies.

5. Interpretations

5.a. Jebel Huwayyah Mound formation

The formation of the JHM began with the deposition in a shallow-
marine setting of the medium-grained lithic sandstone that forms
the matrix of the mound. Soon after this, on a small area of seafloor,
calcite cement precipitated that bound the grains together, forming
a positive mound-shaped structure. Because of the relatively
low concentration of Sr in this cement phase (Fig. 5; online
Supplementary Material Table S2), we infer that the mineralogy
of this early cement phase was originally calcite and not aragonite;
calcite resulting from the recrystallization of aragonite tends to
retain high Sr contents in the order of several thousand ppm
(Buggisch & Krumm, 2005; Peckmann et al. 2007). We interpret
the tubes within the mound to represent conduits through which
the fluids flowed, from which the early-stage calcite cement pre-
cipitated. Proof of upward fluid flow comes from the rotated sand
grains close to the edges of the fluid conduits and winnowed finer
grains from the sediment, leaving only coarse grains in the flat
bases of some of the conduits (Fig. 2a). The rims of inclusion-rich
fibrous calcite (the early cement) in the vugs and tubes precipitated
at the same time as the mound-forming matrix cements.

An alternative explanation for the grain rotation in the tubes
might be the movement of animals through the lithic sediment
to produce burrows prior to cementation. However, burrows tend

Fig. 3. (Colour online) Transverse sections of Jebel Huwayyah Mound fluid conduits. (a, b) Section of one conduit with particularly angular outline in (a) plane polarised light and
(b) cross-polarised light. (c) Conduit with more circular outline. (d) Smaller conduit with very irregular outline. (e, f) Back-scattered electron micrographs of conduit showing late-
stage cement calcite crystals increasing in size to centre of conduit and hollow interior (black space); and, (f) in detail, clear differentiation between inclusion-rich early cement and
inclusion-free late cement. Abbreviations: ec - early cement; lc - late cement. Scale bars: (a–e) = 500 μm; (f) = 100 μm.
to have much more regular shapes (particularly in transverse sections) than the JHM conduits, and are almost invariably sediment filled (e.g. Bromley, 1996). The JHM tubes and vugs are also not post-cementation animal borings, because the cements and grains at the periphery of the conduit walls are not truncated, as can be seen in borings in hardgrounds and reworked cobbles higher up in the Jebel Huwayyah section (Wilson & Taylor, 2001). Plant root traces can be shaped like the JHM tubes, but again, most are filled by later sediments, and may contain carbonaceous remnants (Gregory et al., 2004), which the JHM conduits lack completely.

In addition, animal burrows and plant root traces cannot explain the presence of vugs with the same cement linings as the tubes in the JHM, or the grain winnowing within the conduits. For these reasons we hereafter refer to the JHM tubes as fluid conduits.

Sometime after the formation of the JHM, the second phase of inclusion-free blocky calcite (the late cement) and associated microsparitic cement precipitated in the vugs and fluid conduits from later stage fluids circulating within the mound. The larger size of the late cement crystals compared to the early cement crystals

### Table 1. δ¹⁸O and δ¹³C values from the Jebel Huwayyah Mound (JHM) fluid conduit cements, matrix cements and Qahlah Formation (QF) hardgrounds

| Sample | Description                  | δ¹⁸O (PDB) | δ¹⁸O (SMOW) | δ¹³C (PDB) | Formation temperature (°C) |
|--------|------------------------------|-----------|------------|------------|---------------------------|
| UAE 1-A| Early cement                 | −2.2      | 28.6       | −4.2       | 19                        |
| UAE 2-A| Early cement                 | −3.1      | 27.8       | −3.3       | 24                        |
| UAE 3-A| Early cement                 | −2.6      | 28.3       | −3.6       | 21                        |
| JB4-1  | Early cement                 | −1.3      | 29.6       | −3.9       | 15                        |
| UAE 1-B| Late cement                  | −2.6      | 28.3       | −5.3       | 21                        |
| UAE 2-B| Late cement                  | −3.1      | 27.7       | −7.9       | 24                        |
| UAE 3-B| Late cement                  | −3.0      | 27.9       | −7.2       | 23                        |
| UAE 2-D| Late cement                  | −3.0      | 27.9       | −6.1       | 23                        |
| JB 2-D | Late cement                  | −2.8      | 28.1       | −7.7       | 22                        |
| JB 2-D | Late cement                  | −2.7      | 28.1       | −7.8       | 22                        |
| JB 4-2 | Late cement                  | −1.1      | 29.7       | −7.0       | 14                        |
| JB 4-3 | Late cement                  | −1.4      | 29.4       | −6.0       | 16                        |
| JB 4-4 | Late cement                  | −1.6      | 29.2       | −6.0       | 17                        |
| JB 4-5 | Microsparitic cement         | −1.5      | 29.3       | −5.8       | 16                        |
| UAE 1-C | JHM matrix                  | −3.0      | 27.9       | −3.5       | 23                        |
| UAE 2-C | JHM matrix                  | −2.2      | 28.7       | −3.5       | 19                        |
| UAE 3-C | JHM matrix                  | −2.3      | 28.5       | −3.3       | 20                        |
| JHH 135-A | QF hardground matrix     | −1.3      | 29.6       | −9.1       | 15                        |
| JHH 135-B | QF hardground matrix    | −0.3      | 30.6       | −5.4       | 11                        |
| JHH 135-2A | QF hardground matrix  | −0.7      | 30.2       | −4.8       | 12                        |
| JHH 135-2A | QF hardground matrix  | −0.6      | 30.2       | −4.9       | 12                        |
| JHH 135-2B | QF hardground matrix  | −0.9      | 29.9       | −4.9       | 14                        |
| JHH 135-2B | QF hardground matrix  | −1.2      | 29.7       | −4.9       | 15                        |
| JHH 135-2C | QF hardground matrix  | −4.1      | 26.7       | −7.6       | 29                        |
| JHH 135-2C | QF hardground matrix  | −4.0      | 26.7       | −7.6       | 28                        |
| JHH 146-A | QF hardground matrix  | −0.5      | 30.4       | −6.1       | 11                        |
| JHH 146-B | QF hardground matrix  | −1.5      | 29.3       | −6.3       | 16                        |

Formation temperatures were computed from δ¹⁸O,SMOW values using the calibration by Kim & O’Neill (1997).

**Fig. 4.** Cross plot (δ¹³C v. δ¹⁸O) of Jebel Huwayyah Mound fluid conduit cements, mound matrix and Qahlah Formation hardgrounds. Data from Table 1.
Late Cretaceous seawater compositions (Mg/Ca \begin{footnotesize} Table S1 \end{footnotesize}). These ratios match well with the range of reconstructed around 4 mmol/mol, respectively (online Supplementary Material \begin{footnotesize} Table S2 \end{footnotesize}).

The timing of the late cement formation is difficult to ascertain and most probably occurred after the JHM was covered by a layer of conglomerate and later sediments.

**5.b. Origin and nature of the JHM cement-forming fluids**

Here we use the stable isotope and element composition of the calcite cements in the JHM to reconstruct the origin and nature of the fluids from which they formed, starting with the early cement. The Mg/Ca and Sr/Ca ratios of the reconstructed parent fluid from which this cement precipitated are close or slightly lower than 1000 mmol/mol, and the corresponding Sr/Ca ratios cluster around 4 mmol/mol, respectively (online Supplementary Material \begin{footnotesize} Table S1 \end{footnotesize}). These ratios match well with the range of reconstructed Late Cretaceous seawater compositions (Mg/Ca \begin{footnotesize} = \end{footnotesize} 1000 mmol/mol, Sr/Ca \begin{footnotesize} = \end{footnotesize} 2 to 6 mmol/mol) that are based on calcite veins from ocean crust in the flanks of mid-ocean ridges (Coggon \textit{et al.} 2010; Rausch \textit{et al.} 2013), as well as with theoretical models for the Late Cretaceous (Hardie, 1996; Wallmann, 2001). In contrast, the $\delta^{13}$C values (\begin{footnotesize} −4.2 \% \end{footnotesize} to \begin{footnotesize} −3.3 \% \end{footnotesize}) from the early cement are around 5 \% lower than would be expected for calcite precipitated in equilibrium with dissolved inorganic carbon (DIC) of Cretaceous seawater (\begin{footnotesize} +2 \% \end{footnotesize}; Wilson & Opdyke, 1996; Prokoph \textit{et al.} 2008), and indeed the values recorded from the Maastrichtian Simsiya Formation at Qalhat, NE Oman (0 to \begin{footnotesize} +2 \% \end{footnotesize}; Schlüter \textit{et al.} 2008). An explanation for the negative $\delta^{13}$C values in the JHM early cement requires the mixing of the contemporary seawater DIC pool (modern seawater $\delta^{13}$C\textsubscript{DIC} \begin{footnotesize} = \end{footnotesize} −0 \%), Zeebe & Wolf-Gladrow, 2001) with an isotopically lighter source of dissolved DIC. One common source of DIC with low $\delta^{13}$C values reflecting $^{13}$C depletion is the oxidation of organic carbon (Irwin \textit{et al.} 1977). However, there is no obvious source of significant organic material (e.g. carbon-rich shales) in the sedimentary sequence hosting the JHM, as the Qahlah Formation rests directly on the crystalline rocks of the Semail Ophiolite, so another source of isotopically light DIC needs to be invoked. We suggest that the ophiolite itself could have been this source, through the production of abiogenic methane by serpentinization of peridotite. Methane in modern serpentinite-derived fluids in both the deep-sea and in ophiolite environments has negative $\delta^{13}$C values (\begin{footnotesize} −11.9 \% \end{footnotesize} at Lost City, \begin{footnotesize} −10.3 \% \end{footnotesize} at Logatchev, \begin{footnotesize} −16.7 \% \end{footnotesize} at Rainbow, \begin{footnotesize} −18 \% \end{footnotesize} at Elba and \begin{footnotesize} −7.7 \% \end{footnotesize} in the Zamboales ophiolite; Abrajano \textit{et al.} 1990; Lilley \textit{et al.} 1993; Charlou \textit{et al.} 2002; Proskurowski \textit{et al.} 2008; Meister \textit{et al.} 2018; Sciarra \textit{et al.} 2019). Some of this methane will be oxidized close to the seafloor to produce $^{13}$C-depleted DIC, and the high pH and Ca concentrations in the serpentinite-derived fluids promote carbonate precipitation on mixing with seawater (Palandri & Reed, 2004; Proskurowski \textit{et al.} 2008). These carbonates usually (but not always) have negative $\delta^{13}$C values, often with the same range as the JHM cements. The abundance of serpentinite sand grains in the JHM matrix is ample evidence for serpentinization of the Semail Ophiolite to have occurred in the Jebel Huwayyah area during Late Cretaceous time, certainly prior to the formation of the sand grains, and, as we here suggest, during the deposition of the overlying Qahlah Formation.

The late cement phase precipitated from a fluid with lower Mg/Ca and Sr/Ca ratios than the early cement phase (Fig. 5). Serpentinization fluids show low Mg and high Ca concentrations relative to seawater (e.g. Kelley \textit{et al.} 2001). Moreover, the lower Mg/Ca and Sr/Ca ratios of the JHM late cements are similar to those of serpentinite-hosted calcite from the Logatchev hydrothermal system, which Eickmann \textit{et al.} (2009) interpreted to reflect mixing between seawater and a hydrothermal fluid. Applying this evidence to our model for the JHM, we infer that the fluids from which the late cement phase precipitated comprised a greater proportion of serpentinization-derived fluids to seawater than the early cement phase. This is corroborated by the more negative $\delta^{13}$C values of the late cement (Fig. 4), which indicates a greater influence of methane oxidation on the DIC pool from which this mineral phase precipitated.

Because there is a trend from higher to lower Mg/Ca and Sr/Ca ratios in carbonate cements during diagenesis (e.g. Tucker & Wright 1990; Joseph \textit{et al.} 2013), it is possible that the JHM late cement had a diagenetic origin, from burial diagenesis and/or interaction with meteoric water during early cementation. However, this is rather unlikely for several reasons. First because the early cement and the late cement of the JHM tubes have similar $\delta^{18}$O values (\begin{footnotesize} −3.1 \% \end{footnotesize} and \begin{footnotesize} −1.3 \% \end{footnotesize} versus \begin{footnotesize} −3.1 \% \end{footnotesize} and \begin{footnotesize} −1.1 \% \end{footnotesize}), possibly indicating that the oxygen stable isotope composition of the fluids and formation temperatures did not change much from the time when the early cement formed to the time when the late cement mostly occluded the fluid conduits. Second, we think that later diagenetic alteration would affect both the early and late JMH.
cement phases, leading to homogenization of $\delta^{13}\text{C}$ values, and Mg/Ca and Sr/Ca ratios between the early and later calcite phases. We also note that Schlüter et al. (2008) recorded $\delta^{18}\text{O}$ values from the carbonate-rich Simmsima Formation as low as $-8\,\%\text{o}$, which they interpreted as showing diagenetic alteration; these values are considerably more negative than any found in the JHM cements.

The Mg/Ca and Sr/Ca ratios for the microsparitic cement phase are largely intermediate between the early and late cements (Fig. 5). This may indicate that the microsparitic cement precipitated from a fluid with a composition between that from which the early and late cements precipitated. However, the single $\delta^{13}\text{C}$ value for the microsparitic cement sits in the field of the late-stage cement (Fig. 4), so it is likely the microsparitic cement precipitating fluids were more related to this later phase of cementation in the JHM.

The calculated formation temperatures for the all the JHM cement phases (Table 1) based on their $\delta^{18}\text{O}$ values range between 14 and 24 °C (mean 19.9; $n = 17$), assuming a seawater value of $-1\,\%\text{o}$ for an ice-free Late Cretaceous (Veizer et al. 1997; Prokop et al. 2008). This range is at the lower end or below temperature estimates for Maastrichtian tropical sea-surface temperatures of 20 to 32 °C from planktonic foraminifera (Pearson et al. 2001; Zeebe, 2001), 27 to 32 °C from rudist bivalve aragonite and magnesian calcite cements (Wilson & Opdyke, 1996), and a minimum of 25 °C using the TEX86 proxy (Alsenz et al. 2013). However, these calculated JHM palaeotemperatures should be taken with caution, as we do not know the $\delta^{18}\text{O}$ value(s) for the inferred serpentinitization-derived fluids, nor the degree of mixing with contemporary seawater in the JHM during cement precipitation. Further, the cement $\delta^{18}\text{O}$ values could have been overprinted by later diagenetic events (e.g. Tong et al. 2016), although see the arguments above about a possible diagenetic interpretation.

Support from PAAS-normalized REEs for our interpretation of serpentinitization-derived fluids being involved in the formation of the JHM is equivocal. The REE and Y patterns of the early- and late-stage calcite cements (Fig. 6) show subtle differences between the two generations. The early cement phase displays very uniform patterns with a slight heavy REE (HREE) enrichment. The lack of a positive Y anomaly suggests that the fluid from which the early cement precipitated was modified by water–rock reactions. The late cement does show a positive Y anomaly and has lower HREE concentrations. It is not straightforward to reconcile the differences in the REE patterns with the variability in Mg/Ca ratios and $\delta^{13}\text{C}$ values. This is in part due to the fact that REE and Y systematics of fluids involved in low-temperature serpentinitization have not been investigated to date.

To sum up, we have a number of geochemical lines of evidence that the cements forming the JHM were precipitated from serpentinitization-derived fluids mixed to a greater or lesser extent with Maastrichtian seawater for a period of time during the deposition of the Qahlah Formation. The formation of the JHM tubes is therefore analogous to fluid-induced chimney formation on the modern ocean floor, where the interplay between ascending venting fluids and seawater forms chimneys recording a progressive change from a seawater-like signature in the outermost part to a lesser seawater component in the inner part (Eickmann et al. 2014). Following from this, and with reference to the chemistry of modern serpentinitization-related seeps, we infer that the JMH serpentinitization-derived fluids were alkaline and rich in molecular hydrogen, abiogenic methane and Ca.

5.c. Qahlah Formation hardgrounds

The $\delta^{13}\text{C}$ values of the hardground cements are similar to those of the JHM late cement, although one of the values ($-9.1\,\%\text{o}$) is the most negative of all the carbonates we measured from the Qahlah Formation (Fig. 4), and also more negative than any marine hardgrounds in the geological record documented by Erhardt et al. (2020), with the lowest value of $-5.15\,\%\text{o}$ from the Lower Cretaceous of Oman. From this we infer, again, that the fluid from which the hardground cements precipitated was not pure seawater. Although we have no supporting elemental data, we suggest that this isotopically lighter DIC source was from the same serpentinitization-derived fluid that contributed to the formation of the JHM cements. Some support for this idea is that the hardgrounds are not laterally continuous (i.e. do not represent periods of basin-scale emergence or non-deposition) and are only found in the Jebel Huwayyah section, not in other outcrops of the Qahlah Formation (see Section 2). If this interpretation is correct, then it suggests serpentinitization-derived fluid flow was active for a considerable time during the formation of the Qahlah Formation, but this was localized paleogeographically to the Jebel Huwayyah area. Such localization would be expected given that sites of modern serpentine-related seepage are geographically constrained to small areas of land-based ophiolites and submarine mantle rock exposures (see Section 1).

6. Discussion and conclusions

There are several implications of our interpretation that serpentinitization-derived fluids were involved in the formation of the JHM and conduit-filling cements, and perhaps also the Jebel Huwayyah hardgrounds. First, it expands the temporal duration of serpentinitization-related seepage in the Semail Ophiolite, and the morphological diversity of resulting carbonate mineral structures. Second, it is, to the best of our knowledge, the only ancient example of a carbonate mineralized seafloor feature formed from serpentinitization-related seepage from shallow-marine settings.

Today, the Semail Ophiolite is host to numerous terrestrial hyperalkaline springs, the famous Omani ‘Blue Pools’ (e.g. Neal & Stanger, 1984; Chavagnac et al. 2013; Giampouras et al. 2020), where serpentinitization-derived fluids mix with meteoric waters. Earlier evidence of serpentinitization of the Semail Ophiolite comes from Wadi Fins in SE Oman where there are veins of calcite and subordinate dolomite within a host rock of serpentinitized peridotite, associated with deep clastic dykes filled with sedimentary carbonates (de Oses & Kelemen, 2018, 2020; Cooperdock et al. 2020). These calcite veins have $^{87}\text{Sr}/^{86}\text{Sr}$ ratios indicative of an age around the Cretaceous–Paleocene boundary (de Oses & Kelemen, 2018) and (U–Th)/He values in hydrothermal magnetite in the veins, which give an age estimate of 15 ± 4 Ma, equating to the Miocene (Cooperdock et al. 2020). Cooperdock et al. (2020) interpreted these calcite veins as having formed over a considerable period of time through the interaction of the Semail Ophiolite mantle peridotites with pore fluids derived from overlying Cretaceous and Paleocene limestone formations. The JHM shows that similar processes started earlier, possibly in late Campanian to early Maastrichtian time, soon after the end of ophiolite obduction onto the Arabian continental margin (Searle & Cox, 1999). Post-emplacement serpentinitization of the Semail Ophiolite likely occurred even before this, as lateritic debris in conglomerates at the base of Maastrichtian marine sediments show that parts of the ophiolite were raised above sea-level prior to the early Maastrichtian transgression and subjected to sub-aerial weathering.
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(Coleman, 1981; Al-Khirbash, 2015). It seems likely that some serpentinization could have occurred during this weathering phase. Further, based on δ18O data Gregory & Taylor (1981) suggested that some serpentinization of the Semail Ophiolite mantle sequence was taking place even as it was being formed. The reconstructed composition of the JHM fluids differs from the modern Semail Ophiolite hyperalkaline springs in being a mixture of serpentinization-derived fluids with seawater rather than with meteoric waters and the precipitation of calcite only, rather than calcite, aragonite and brucite. Because brucite is retro-soluble, it is conceivable that this mineral was precipitated during the formation of the JHM and was then replaced during diagenesis. However, there is no petrological evidence (e.g. pseudomorphic minerals) in the studied JHM thin-sections for this sort of replacement process. Further, brucite has not been recorded in the Wadi Fins calcite veins, which de Obeso & Kelemen (2018) interpreted as an indication of non-isochronous serpentinization at that site. The JHM calcite cement δ13C values are more negative than those from the Wadi Fins calcite veins (−1.3 to +0.6‰), which correspond rather to the values from the overlying Simsima Formation (de Obeso & Kelemen, 2018).

Close modern analogues to the environment in which the JHM formed are the shallow-marine serpentinization-derived fluid and gas seeps found in the Bay of Prony, New Caledonia and off the island of Elba, Italy. The Pryon hydrothermal field (PHF) comprises a suite of fluid seeps in a marine lagoon, in the intertidal zone and onshore springs (Launay & Fontes, 1985; Monnin et al. 2014; Pisapia et al. 2017). The marine seeps are precipitating submarine carbonate structures with complex morphologies in the 30 m and 50 m depth range, including the well-known Aiguille de Prony, which is 35 m tall and reaches to within 2 m of the water surface. These structures are highly porous and are formed largely of calcite, with increasing amounts of brucite mixed with Mg-carbonates and aragonite towards their interiors (Pisapia et al. 2017). The PHF fluids have a high pH and high concentration of aqueous calcium and hydroxide, with only traces of other solutes (Monnin et al. 2014). The low salinity values agree with fluids derived from the mixing of meteoric water with fluids derived from serpentinization, even though in most places in the Bay of Prony fluids are now discharging into the sea and ultimately mixing with seawater. In this respect they differ from the composition of the fluids which formed the JHM, which we infer to have had a large seawater component, at least for the early cement phase.

Another analogous modern setting to the JHM are sites offshore Elba in the Tyrrhenian Sea (Meister et al. 2018; Sciarra et al. 2019). At the Pomonte site, gas emission occurs across an area of 1000 m² of seafloor in 10–13 m water depth through quartz-rich, organic-poor sands, deposited on top of rocks of the Ligurian Ophiolite, which locally include serpentinitized peridotite. Gas bubbles are enriched in methane (δ13Cmethane ~ −18‰) and hydrogen, and have a very low CO2 content. The gas seeps are associated with areas of discoloured sediment containing semi-lithified carbonate crusts, at between 20 and 40 cm depth within the sediment. The crusts are formed of spherulitic, fibrous aragonite that cements sand grains and sometimes bryozoans and seagrass rhizome fibres. The crusts have δ13C values of between −17 and +2‰ and δ18O values of approximately +1.5‰. Based on the decrease in pore water sulphate concentrations at the gas seeps relative to seawater, increases in sulphide and DIC, and the negative δ13C values, Meister et al. (2018) inferred that the Elba seep carbonate cements are the product of microbial sulphate-dependent anaerobic oxidation of abiotic methane (AOM).

This scenario of AOM-related carbonate precipitation in shallow-marine sediments can be applied to the hardgrounds in the Qahlah Formation, although the δ13C values of the latter are more positive than some of the Elba seep carbonate values. The JHM itself contrasts with the Elba example because the carbonates are calcite rather than aragonite, which may reflect the predominantly calcite precipitation in Cretaceous seas (Sandberg, 1983), rather than any fundamental differences in seepage fluid composition. Other differences are that the JHM shows evidence of several stages of fluid mixing and cement precipitation to form seafloor rather than subsurface features as in Elba. Nonetheless, there are enough similarities between the New Caledonian and Elba examples and the JHM to indicate the latter (and possibly the Qahlah Formation hardgrounds) is another example of shallow-marine serpentine-derived seepage, and therefore adds a data point in the geological record to our knowledge of this interesting phenomenon.

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References

Abbasi IA, Heersi OS and Al-Harthi A (2014) Late Cretaceous conglomerates of the Qahlah Formation, north Oman. In Tectonic Evolution of the Oman Mountains (eds HR Rollinson, MP Searle, IA Abbasi, A Al-Lazki and MH Al Kindi), pp. 325–41. Geological Society of London, Special Publication no. 392.

Abdelghany O (2006) Early Maastrichtian larger foraminifera of the Qahlah Formation, United Arab Emirates and Sultanate of Oman borders. Cretaceous Research 27, 898–906.

Abrajano TA, Sturchio NC, Bohlke JK, Lyon GL, Poreda RJ and Stevens CM (1988) Methane-hydrogen gas seeps, Zambales ophiolite, Philippines: deep or shallow origin? Chemical Geology 71, 211–22.

Abrajano TA, Sturchio NC, Kennedy BM, Lyon GL, Muehlenbachs K and Bohlke JK (1990) Geochemistry of reduced gas related to serpentinization of the Zambales ophiolites, Philippines. Applied Geochemistry 5, 625–30.

Agrinier P, Cornen G and Beslier M-O (1996) Mineralogical and oxygen isotopic features of serpentinites recovered from the ocean/continent transition in the Iberia abyssal plain. In Proceedings of the Ocean Drilling Program, Scientific Results, vol. 149 (eds RB Whitmarsh, DS Sawyer, A Klaus and DG Masson), pp. 541–52. College Station, Texas.

Al-Khirbash S (2015) Genesis and mineralogical classification of Ni-laterites, Oman Mountains. Ore Geology Reviews 65, 199–212.

Alsenz H, Regnery J, Ashckenazi-Polivoda S, Meilijson A, Ron-Yankovich L, Abramovich S, Illner P, Almogi-Labin A, Feinstein S, Berner Z and Püttmann W (2013) Sea surface temperature record of a Late Cretaceous tropical southern Tethys upwelling system. Palaeogeography, Palaeoclimatology, Palaeoecology 392, 350–8.

Alsharan AS and Nasir SJY (2017) Sedimentological and geochemical interpretation of a transgressive sequence: the Late Cretaceous Qahlah Formation in the western Oman Mountains, United Arab Emirates. Sedimentary Geology 101, 227–42.

Bach W, Pauliuk H, Garrido CJ, Ildefonse B, Meurer WP and Humphris SE (2006) Unraveling the sequence of serpentinization reactions: petrography, mineral chemistry, and petrography of serpentinites from MAR 15°N (ODP...
in alkaline springs, Samal Ophiolite, Oman. Chemical Geology 533, 119435. doi: 10.1016/j.chemgeo.2019.119435.

Glenie KW, Boeuf MGA, Clarke MH, Moody-Stuart M, Pilaar WFH and Reinhardt BM (1973) Late Cretaceous nappes in Oman Mountains and their geologic evolution. American Association of Petroleum Geologists Bulletin 57, 5–27.

Gregory MR, Martin AJ and Campbell KA (2004) Compound trace fossils formed by plant and animal interactions: Quaternary of northern New Zealand and Sapelo Island, Georgia (USA). Fossils and Strata 51, 88–105.

Gregory RT and Taylor HP (1981) An oxygen isotope profile in a section of Cretaceous oceanic crust, Samal ophiolite, Oman: evidence for δ^{18}O buffering of the oceans by deep (>5 km) seawater–hydrothermal circulation at mid-ocean ridges. Journal of Geophysical Research 86, 2737–55.

Hardie LA (1996) Secular variations in seawater chemistry: an explanation for the coupled secular variation in the mineralizations of marine limestones and potash evaporates over the past 600 m.y. Geology 24, 279–83.

Irwin H, Curtis C and Coleman M (1977) Isotopic evidence for source of diagenetic carbonates formed during burial of organic-rich sediments. Nature 269, 209–13.

Joseph C, Campbell KA, Torres ME, Martin RA, Pohlman JW, Riedel M and Rose K (2013) Methane-derived authigenic carbonates from modern and ancient carbonate environments. Journal of Geophysical Research: Solid Earth 118, 3455–58.

Klein F, Humphris SE, Guo W, Schubotz F, Schwarzenbach EM and Orsi WD (2015) Fluid mixing and the deep biosphere of a fossil Lost City-type hydrothermal system at the Iberia Margin. Proceedings of the National Academy of Sciences of the United States of America 112, 12036–41.

Lafay R, Baumgartner LP, Schwartz S, Picazo S, Montes-Hernandez G and Vennemann T (2017) Petrologic and stable isotopic studies of a fossil hydrothermal system in ultramafic environment (Chenai belt, Western Alps, France): processes of carbonate cementation. Lithos 294–295, 319–38.

Ludwig KL, Butterfield DA, Olson EJ, Lupton JE, Macko SA and McDuff RE (2010) Elevated concentrations of formate, acetate and dissolved organic carbon found at the Lost City hydrothermal field. Geochimica et Cosmochimica Acta 74, 941–52.

Meister P, Wiedling J, Lott C, Bach W, Kühfuß H, Wegener G, Böttcher ME, Deussen C, Lichtschlag A, Bernasconi SM and Weber M (2018) Anaerobic methane oxidation inducing carbonate precipitation at abiotic methane seeps in the Tuscan archipelago (Italy). PLoS ONE 13, e0207305. doi: 10.1371/journal.pone.0207305.

Mininni C, Chałavagnc V, Boulart C, Ménez B, Gérard M, Gérard E, Pisapia P, Lavoie D and Chi G (2020) On the controls of mineral assemblages and textures of the mid-ocean ridge. Biogeosciences 17, 83–97.

Palandrini JL and Reed MH (2004) Geochemical models of metasomatism in ultramafic systems: serpentinization, oxidization, and sea floor carbonate chimney precipitation. Geochimica et Cosmochimica Acta 68, 1115–33.
Cretaceous serpentinite seep

Pearce NJG, Perkins WT, Westgate JA, Gorton MP, Jackson SE, Neal CR and Chenery SP (1997) A compilation of new and published major and trace element data for NIST SRM 610 and NIST SRM 612 Glass Reference Materials. Geostandards Newsletter 21, 115–44.

Pearson PN, Ditchfield PW, Singano J, Harcourt-Brown KG, Nicholas CJ, Olsson RK, Shackleton NJ and Hall MA (2001) Warm tropical sea surface temperatures in the Late Cretaceous and Eocene epochs. Nature 413, 481–7.

Peckmann J, Campbell KA, Walliser OH and Reitner J (2007) A Late Devonian hydrocarbon-seep deposit dominated by dimereloid brachiopods, Morocco. Palaeoices 22, 114–22.

Pisapia C, Gérard E, Gérard M, Lecourt L, Lang SQ, Pelletier B, Payri CE, Monnin C, Guentas L, Postec A, Quéméneur M, Erauso G and Ménez B (2017) Mineralizing filamentous bacteria from the Prony Bay Hydrothermal Field give new insights into the functioning of serpentinization-based subsea-floor ecosystems. Frontiers in Microbiology 8, 57. doi: 10.3389/fmicb.2017.00057.

Prokoph A, Shields GA and Veizer J (2008) Compilation and time-series analysis of a marine carbonate δ18O, δ13C, 87Sr/86Sr and δ34S database through Earth history. Earth-Science Reviews 87, 113–33.

Proskurowski G, Lilley MD, Seewald JS, Früh-Green GL, Olson EJ, Lupton JE, Sylva SP and Kelley DS (2008) Abiogenic hydrocarbon production at Lost City Hydrothermal Field. Science 319, 604–7.

Rauk J, Bohn F, Bach W, Klügel A and Eisenhauer A (2013) Calcium carbonate veins in ocean crust record a threefold increase of seawater Mg/Ca in the past 30 million years. Earth and Planetary Science Letters 362, 215–24.

Rimstidt JD, Balog A and Webb J (1998) Distribution of trace elements between carbonate minerals and aqueous solutions. Geochimica et Cosmochimica Acta 62, 1851–63.

Sandberg PA (1983) An oscillating trend in Phanerozoic non-skeletal carbonate mineralogy. Nature 305, 19–22.

Schlüter M, Steuber T, Parente M and Mutterlose J (2008) Evolution of a Maastrichtian–Paleocene tropical shallow-water carbonate platform (Qalhat, NE Oman). Facies 54, 513–27.

Sciarrà A, Saroni A, Etiöpe G, Coltorti M, Mazzarini F, Lott C, Grassa F and Italiano F (2019) Shallow submarine seep of abiotic methane from serpentinized peridotite off the Island of Elba, Italy. Applied Geochemistry 100, 1–7.

Searle M and Cox J (1999) Tectonic setting, origin and obduction of the Oman ophiolite. Geological Society of America Bulletin 111, 102–22.

Skelton PW, Nolan SC and Scott RW (1990) The Maastrichtian transgression onto the northwestern flank of the Proto-Oman Mountains: sequences of rudist-bearing beach to open shelf facies. In The Geology and Tectonics of the Oman Region (eds AHF Robertson, MP Searle and AC Ries), pp. 521–47. Geological Society of London, Special Publication no. 49.

Sleep NH, Meibom A, Frödriksson T, Coleman RG and Bird DK (2004) H2-rich fluids from serpentinization: geochemical and biotic implications. Proceedings of the National Academy of Sciences of the United States of America 101, 12818–23.

Smith AB, Morris NJ, Gale AS and Kennedy WJ (1995) Late Cretaceous carbonate platform faunas of the United Arab Emirates–Oman border region. Bulletin of The Natural History Museum, London, Geology Series 51, 91–119.

Taylor SR and McLennan SM (1985) The Continental Crust: Its Composition and Evolution. Oxford: Blackwell Scientific Publications, 312 pp.

Tong H, Wang Q, Peckmann J, Cao Y, Chen L, Zhou W and Chen D (2016) Diagenetic alteration affecting δ18O, δ13C and 87Sr/86Sr signatures of carbonates: a case study on Cretaceous seep deposits from Yarlung-Zangbo Suture Zone, Tibet, China. Chemical Geology 444, 71–82.

Tucker ME and Wright PV (1990) Carbonate Sedimentology. Oxford: Blackwell Science.

Veizer J, Bruckschen P, Pawelik F, Diener A, Podlaha OG, Carden GAF, Jasper T, Korte C, Strauss H, Azmy K and Ala D (1997) Oxygen isotope evolution of Phanerozoic seawater. Palaeogeography, Palaeoclimatology, Palaeoecology 132, 159–72.

Wallmann K (2001) Controls on the Cretaceous and Cenozoic evolution of seawater composition, atmospheric CO2 and climate. Geochimica et Cosmochimica Acta 65, 3005–25.

Wilson PA and Opdyke BN (1966) Equatorial sea-surface temperatures for the Maastrichtian revealed through remarkable preservation of metastable carbonate. Geology 24, 555–8.

Wilson MA and Taylor PD (2001) Palaeoecology of hard substrate faunas from the Cretaceous Qahlah Formation of the Oman Mountains. Palaeontology 44, 21–41.

Zeebe RE (2001) Seawater pH and isotopic paleotemperatures of Cretaceous oceans. Palaeoecography, Palaeoclimatology, Palaeoecology 170, 49–57.

Zeebe RE and Wolf-Gladrow D (2001) CO2 in Seawater: Equilibrium, Kinetics, Isotopes. Elsevier Oceanography Series 65. Amsterdam: Elsevier.