QUALITY EVALUATION OF HARDY KIWIFRUIT (ACTINIDIA KOLOMIKTA) USING NON-DESTRUCTIVE AND HOLISTIC RESEARCH METHODS

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Fresh hardy kiwifruit (Actinidia kolomikta (Maxim. & Rupr.) Maxim.) possible to store for only two weeks at 0–5°C, therefore in order to maintain longer, the fruit have to be processed. The purpose of the study was to determine the quality of the Actinidia kolomikta fruit using research methods that are faster and easier, i.e. bioelectric Vincent method and biocrystallization. Four cultivars of Actinidia kolomikta fruits – ‘Landè’, ‘Paukštès Šakarva’, ‘Laiba’ and ‘Lankè’ – were investigated. Fresh, frozen and freeze-dried fruit were analysed. The amount of dry matter was determined by drying the samples to constant mass at 105 °C, ascorbic acid was determined by titration with 2,6-dichlorphenol-indophenol sodium salt dehydrate. The pH and redox potential were measured by 781 pH/Ion Meter, electrical conductivity was measured by conductometer. P value as combined parameter was calculated according to the formula. Biocrystallization studies were carried out with fresh, frozen and freeze-dried fruits of all cultivars. The images derived from the encoded by sort samples were characterised with respect to the visual strength of form expression and were described by 10 criteria, criteria were evaluated using a 5-point scale. Research results showed that the dry matter and ascorbic acid content of A. kolomikta fruit significantly depended on the cultivar. Dry matter content of frozen fruit decreased from 4 to 7%, but increased during freeze-drying process by 6 to 7 times compared to the fresh fruit. Fruit processing methods increased pH, redox potential and P values. The lowest redox potential and P values were determined for fresh fruit, medium for freeze-dried and the highest for frozen fruit samples.

Keywords: ascorbic acid, biocrystallization, electrochemical parameters.

INTRODUCTION

Hardy kiwifruit Actinidia kolomikta (Maxim. & Rupr.) Maxim. are evaluated as a healthy fruit because of high content of biologically active substances. A. kolomikta fruit of different cultivars accumulate large amounts of vitamin C that varies from 2423 to 11460 mg kg⁻¹ (Pranckietis et al., 2009; Paulauskiene et al., 2013). The ripe fruits fall down; therefore, all fruits are harvested at the same time, when most of them are unripe. Paulauskië et al. (2013) previous studies have indicated that the greatest amounts of vitamin C are found in the unripe A. kolomikta fruit.

Fresh A. kolomikta fruit possible to store for only 10–14 days at 0–5°C, therefore in order to maintain longer, the fruit have to be processed.

The oldest and most frequently used method of preservation is food freezing, which preserves foods taste, texture, and nutritional value better compared to other methods. In low temperatures microorganisms cannot multiply, chemical reactions decreases, and cells metabolic reactions delays (Delgado, Sun, 2000). The bioactive compounds of frozen products are better protected than using other storage methods (Tosun, Yucecan, 2008). Freeze-drying is treated as one of the most progressive methods for drying high-value products because shrinkage is avoided and produces materials with superior flavour, aroma, colour retention and unchanged nutritional quality (Oikonomopoulou et al., 2011). The nutrient content is reduced only during the sublimation process when the water from the products evaporates. Oikonomopoulou et al. (2011) states, that 98% of all nutrients in freeze-dried products upheld. Another advantage, that low water activity virtually removes microbiological hazards.

Fruit quality is usually determined by its chemical composition and some physical characteristics. Conventional methods used by researchers are often complicated, require expensive reagents, equipment and are prolonged. Therefore, researchers are looking for faster and easier methods to determine fruit quality. According to Bloksma et al. (2001), Bioelectric Vincent method is one of the promising novel fruit quality assessment methods, which provide more knowledge about metabolism and physiological processes. Gajewski et al. (2007) states that life processes in plants can be described as chains of electro-chemical or redox reactions gain from the activity of electrons. Three basic factors: pH, redox potential (rH in mV) and resistivity (R in Ω), make up the basis of the method, which translated into P-value an electrochemical parameter of product quality (Kappert, Meltsch, 2007). According to Bioelectric Vincent method, better fresh or processed product
quality is attained by a low redox potential and P-value, but a higher resistivity (Wolf, Rey, 1997). Bioelectric Vincent method has been used for the quality assessment of a few fruits and vegetables and found effective including apples, oranges, strawberries, pumpkins, carrots, and tomatoes (Ergun, Jezik, 2011). Collection of biocrystallization data have motivate this method as a useful scientific research for food quality analysis. This holistic method is based on the chemical and physical reaction of food ingredients with a solution of copper chloride (CuCl₂·2H₂O) (Fritz et al., 2017). The analysis is performed on glass plates and specific dendritic crystallisation images are obtained. The images form through a self-organization process which is influenced by the ingredients (Kokornaczyk et al. 2011; Busscher et al., 2014) and seams to reflect physiological processes of fruit or vegetables like ripening, decomposition and etc. (Fritz et al. 2011). Laboratory procedures of biocrystallization have been standardized (Busscher et al., 2010; Huber et al., 2010; Kahl et al., 2015). The properties of the pattern or picture can be evaluated by human or using special computer programme (Huber et al., 2010; Szulc et al., 2010; Doesburg, Nierop, 2013). Several scientific researches refer that biocrystallization method is particularly suitable for verifying the authenticity of organic products (Siderer et al., 2005; Szulc et al., 2010). The evaluation of the image is based on (1) comparing the strength of the form expression, measured by the concentration of the sample in the image required to form specific structures of the product in reaction with the copper chloride; and (2) linking samples consisting of under-studied images to reference images with varying degrees of maturity or degradation (Fritz et al., 2011; Doesburg et al., 2015).

The purpose of the study was to determine the quality of the Actinidia kolomikta fruit using standard and unconventional research methods.

RESEARCH METHODS

Fruit of Actinidia kolomikta cultivars ‘Laiba’, ‘Lankė’, ‘Landė’ and ‘Paukštės Šakarva’ were collected from the experimental orchard (54°53’ N, 23°50’ E) of the Aleksandras Stulginskis University in the central region of Lithuania. The fruit were handpicked after 62 days after the most abundant flowering at the beginning of August in 2016 and 2017. The fruit were put into perforated boxes and transported to the laboratory. 1 kg of fruit of each cultivar was immediately analysed. The fruit in the boxes, 4 x 1.0 kg for each cultivar, were placed in the freezer (Elcold, Denmark) and frozen at -34°C. The fruit were analysed after eight hours of freezing. The fruit for freeze drying were cut in half. Freeze drying of 5 kg of each cultivar fruit were carried out in a freeze dryer (Sublimator 3x4x5, Zirbus GmbH, Germany). The freeze drying temperature was -40°C and condenser temperature -72 ± 1°C with chamber pressure of 10 Pa. Secondary heating was carried up to 25°C. The freeze drying was completed in 48 hours. The freeze dried fruit were powdered in ultra-centrifugal mill (ZM 200, Retsch GmbH, Germany). The fruit powder samples until analysed were stored in sealed containers.

Frost, frozen and freeze-dried fruit were analysed. For chemical analysis the whole fresh and unfreeze fruit were homogenised with peers and seeds. The amount of dried matter was determined by drying the samples to constant mass at 105 °C, ascorbic acid was determined by titration with 2,6-dichlorophenol-indophenol sodium salt dehydrate (LST ISO 6557-2:2000).

The electrochemical parameters were determined in homogenized fresh and unfreeze fruit samples. Freeze-dried fruit powder mixed with distilled water to homogenize fruit consistency. pH and redox potential (rH) were measured by 781 pH/Ion Meter (Metrohm, Switzerland), electrical conductivity (electrical conductivity is the reciprocal of electrical resistivity) – by conductometer inoLab Cond 7310 (WTW, Germany). P value as combined parameter of three mentioned parameters was calculated according to the formula: \( P = \frac{29.07 \cdot (rH - 2pH)^2 \cdot rH_{10}^{-1}}{\mu W} \), where: rH – redox potential (mV); pH – the hydrogen-ion activity; rHo – recalculated specific electrical conductivity (µS cm⁻¹) (Meier-Ploeger, Vogtmann, 1991).

For the crystallization method, 2 g of frozen homogenized fruit and 0.5 g of freeze-dried fruit powder and 100 ml distilled water were first mixed, extracted 30 min. and then filtered through No. 604 filter paper. Floatglass plates of 2 mm thickness and a surface area of 10.5 x 10.5 cm were used. Plexiglas rings (diameter = 9 cm; height = 1 cm) were mounted with paraffin on the glass plates. 7.5 ml of juice mixture, 7.5 ml 10 % CuCl₂ and 15 ml of distilled water were pipetted into the resulting dish and crystallised in a crystallisation chamber (1.9 x 1.5 x 1.3 cm) at 30 °C with 50 % humidity. Three dishes per sample (36 dishes per series) were placed into one chamber. The chamber used was roughly comparable to that described by Kahl (2007), differing in the dimensions and equipment. The images derived from the encoded by sort samples were characterised with respect to the visual strength of form expression by comparing image features between the different fruit cultivars. The images were described by 10 criteria: abundance of branches, central arrangement of branches, branching intensity, regularity of branching, brightness of branching, abundance of branching fill, length of branches (spiral, linear, parabolic), radial density of branches, smoothness of branches. Criteria were evaluated using a 5-point scale: 1 point was given for the weakest visual feature and 5 points was given for the strongest visual feature.

Statistical analysis was carried out with software TIBCO Statistica, version 7 (TIBCO Software, USA). The results were analysed using factorial analysis of variance (ANOVA). Differences between fruit cultivars and processing methods were analysed. The arithmetical means and standard deviations (SD) of the experimental data were calculated. Fisher’s least-significant-difference (LSD) test was applied to the experimental results to assess differences between mean values at the significance level of \( P < 0.05 \). Correlation and regression analyses were performed to determine the strength and character of the relationships between variables.

RESULTS AND DISCUSSION

The accumulation of bioactive components in plants varies depending on cultivar, variety, geographic or meteorological factors, agricultural practices and soil composition. In the case of A. kolomikta fruit, some of the biochemical compounds significantly depend on the cultivar (Paulauskiene et al., 2013). Significant differences were
observed in the dry matter and ascorbic acid contents of fruit after harvest (Table 1). The greatest amount of dry matter accumulated in 'Laiba' fruit (Table 1). Dry matter contents of frozen fruit were less from 4 to 7% compared to the fresh fruit. Statistical analysis of data shows significant differences among all cultivars. During the freeze-drying process the dry matter content of fruit increased by 6 to 7 times compared to the fresh fruit.

The greatest amount of ascorbic acid was observed in 'Landė' fruit. Our previous studies indicate that fruit of this cultivar usually accumulate larger amounts of ascorbic acid (Paulauskiene et al., 2015). Delgado and Sun (2000) indicated that amount of ascorbic acid in frozen fruit is most influenced by pre-freezing operations and freezing process basely has no significant effect on vitamin loss. In our case freezing process reduced ascorbic acid content 1.2 times and freeze-drying 2.0 times in fruit of all cultivars compare to fresh fruit. A significant positive correlation was observed between ascorbic acid and cultivar (\(r = 0.350, R^2 = 0.123\)), but negative correlation was recorded between ascorbic acid and fruit processing method (\(r = -0.662, R^2 = 0.438\)).

The pH of A. kolomikta fresh and processed fruit ranged from 2.84 to 3.86 (Table 2). Fruit processing increased pH values. The pH of the frozen fruit varied significantly from fresh fruit, but differences between frozen and freeze-dried fruit were insignificant. According to Gajewski et al. (2007), pH value indicates the ions acidity level and externalizes energetic aspects of life process. The pH rise means a loss of fruit vitality (Danilcenko et al., 2005).

The redox potential is of central interest for electrochemical research because it represents the intensity of oxidation-reduction reactions. At \(rH < 28.3\) – are reducing systems, that can release electrons to other systems with lower \(rH\), at \(rH > 28.3\) – are oxidant systems that can accept electrons from systems with higher \(rH\) (Garban, 2008). The values of redox potential for the tested fruit samples were higher than 28 mV, which means the domination of oxidative environment (Table 2). The redox potential for the tested fruit varied from 74.70 mV in fresh ‘Laiba’ fruit to 139.70 mV in frozen ‘Laiba’ fruit. The redox potential values of frozen fruit were higher and the lowest values were established for fresh fruit. That mean plant cells can use free enthalpy for their activity, and fresh fruit are more suitable to the human organism (Paulauskiene et al., 2006).

Table 1. Chemical composition of Actinidia kolomikta fruit

| Cultivar/Fruit processing method | Fresh | Frozen | Freeze-dried |
|---------------------------------|-------|--------|--------------|
|                                 | Dry matter % |       | Vitamin C mg 100g (DW) |
| 'Paukštė Šakarva'                | 14.56±0.18\(^a\) | 13.67±0.09\(^b\) | 94.11±0.13\(^c\) |
| 'Landė'                         | 15.12±0.24\(^a\) | 14.56±0.31\(^b\) | 94.39±0.05\(^c\) |
| 'Laiba'                         | 13.11±0.35\(^a\) | 12.18±0.03\(^b\) | 93.53±0.38\(^c\) |
|                                 | 15.93±0.27\(^a\) | 15.25±0.12\(^b\) | 94.60±0.15\(^c\) |

Table 2. Electrochemical characteristic of Actinidia kolomikta fruit

| Cultivar/Fruit processing method | Fresh | Frozen | Freeze-dried |
|---------------------------------|-------|--------|--------------|
|                                 | pH    | Redox potential mV | P value µW |
| 'Paukštė Šakarva'                | 3.11±0.02\(^a\) | 3.32±0.02\(^a\) | 3.32±0.01\(^a\) |
| 'Landė'                         | 3.08±0.06\(^a\) | 3.19±0.01\(^a\) | 3.23±0.02\(^a\) |
| 'Laiba'                         | 2.84±0.06\(^a\) | 3.24±0.02\(^a\) | 3.26±0.01\(^a\) |
| mean                            | 3.34±0.07\(^a\) | 3.65±0.03\(^a\) | 3.66±0.02\(^a\) |
|                                 | 3.99 | 3.35 | 3.42 |
| 'Paukštė Šakarva'                | 78.40±4.93\(^h\) | 131.10±2.69\(^h\) | 82.20±5.16\(^h\) |
| 'Landė'                         | 90.00±7.24\(^i\) | 118.70±3.09\(^i\) | 92.30±6.04\(^i\) |
| 'Laiba'                         | 105.10±8.40\(^j\) | 120.30±2.84\(^j\) | 123.90±3.38\(^j\) |
| mean                            | 74.70±7.80\(^j\) | 139.70±3.95\(^j\) | 89.90±4.70\(^j\) |
|                                 | 87.05 | 127.45 | 97.08 |
| 'Paukštė Šakarva'                | 14.75±0.40\(^b\) | 37.39±0.45\(^c\) | 20.66±0.60\(^b\) |
| 'Landė'                         | 17.30±0.23\(^b\) | 33.69±0.53\(^b\) | 22.41±0.78\(^b\) |
| 'Laiba'                         | 14.14±0.17\(^a\) | 34.68±0.57\(^c\) | 25.35±0.81\(^b\) |
| mean                            | 15.53±0.38\(^c\) | 32.68±0.39\(^c\) | 23.65±0.80\(^b\) |

P-value is used to define vitality of the organism and energy distribution tendencies (Bloksma et al., 2001). The lowest P-values were determined for fresh fruit, medium for freeze-dried and the highest for frozen fruit samples (Table 2). The P-value of the fresh 'Landė' fruit was the lowest. According to Gajewski et al. (2007), the lower P-value means that product is more suitable as food and there is healthier and nutritious.
Correlation analysis results shown in Table 3. A weak negative correlation between cultivar and fruit pH was established (Table 3). It was found, that the processing method influenced fruit electrochemical parameters, and correlated most strongly with fruit pH. The strongest positive correlation was fixed between redox potential and P-value.

Table 3. Correlation coefficients ($P < 0.01$)

|                        | pH                  | pH                  | P                   |
|------------------------|---------------------|---------------------|---------------------|
| Cultivar               | $r = -0.334$, $R^2 = 0.112$ | -                   | -                   |
| Processing method      | $r = 0.525$, $R^2 = 0.276$ | $r = 0.189$, $R^2 = 0.035$ | $r = 0.384$, $R^2 = 0.148$ |
| P (fresh fruit)        | $r = 0.361$, $R^2 = 0.130$ | $r = 0.799$, $R^2 = 0.634$ | -                   |
| P (frozen fruit)       | $r = 0.340$, $R^2 = 0.116$ | -                   | -                   |
| P (freeze-dried fruit) | $r = -0.398$, $R^2 = 0.158$ | -                   | -                   |

Visual evaluation of crystallization images of fresh, frozen and freeze-dried A. kolomikta fruit were performed primarily by estimating impression of the typical character, which made the whole picture, and then by differentiation the separate morphological features. Crystallization images of fresh fruit are shown in Figure 1.

All criteria for the crystallization images were evaluated using a 5-point scale and the obtained scores were sum up. Freeze-dried fruit crystallograms were assessed on the highest scores, sum of points ranged from 19.5 for ‘Paukštės Šakarva’ to 36.5 for ‘Lankė’ (Figure 2). Fresh fruit crystallograms were rated on similar score, and sum of points ranged from 18.5 for ‘Lankė’ to 34.5 for ‘Paukštės Šakarva’ fruit. The lowest sum of points was intended for frozen fruit and ranged from 16.5 for ‘Landė’ to 29.5 for ‘Lankė’. This sum characterizes energy value of the product. The higher the value of the product, the more it is suitable for human consumption. The highest value was identified for ‘Lankė’ fruit.
CONCLUSIONS

Dry matter and ascorbic acid content of A. kolomikta fruit significantly depended on the cultivar. Dry matter contents of frozen fruit decreased from 4 to 7%, but increased during freeze-drying process by 6 to 7 times compared to the fresh fruit.

Fruit processing methods increased pH, redox potential and P values. The lowest redox potential and P-values were determined for fresh fruit, medium for freeze-dried and the highest for frozen fruit samples.

Freeze-dried fruit crystallograms were assessed on the highest scores and on the similar score were rated fresh fruit crystallograms. The results of the crystallograms evaluation confirm the results obtained from the estimation of the electrochemical parameters.

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