Phenolic Compounds from the Leaves of Castanopsis fargesii

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Abstract: In the course of a phytochemical and chemotaxonomical investigation of Castanopsis species (Fagaceae), three new phenolic compounds, (3R,1’S)-[1’S-(6’-O-galloyl)-β-D-glucopyranosyl]oxyethyl-3-hydroxy-dihydrofuran-2(3H)-one (1), (2R,3S)-2-[2’-(galloyl)oxyethyl]-dihydroxybutanoic acid (2), and (3S,4S)-3-hydroxymethyl-3,4-dihydro-5,6,7-trihydroxy-4-(4’-hydroxy-3’-methoxyphenyl)-1H-[2]-benzopyran-1-one (3) were isolated from the fresh leaves of Castanopsis fargesii. In addition, a known phenolic glycoside, gentisic acid 5-O-α-L-rhamnopyranosyl-(1→2)-β-D-glucopyranoside (4) was also isolated and identified. Their structures were elucidated by means of spectroscopic techniques including one- and two-dimensional NMR techniques.

Keywords: Castanopsis fargesii; Castanopsis; phenolic; cytotoxicity

1. Introduction

The genus Castanopsis belongs to the Fagaceae family and is commonly found in the evergreen forests of East Asia. There are about 120 species of Castanopsis, however, the classical plant taxonomy of the species is very complicated and sometimes confusing [1], thus, the application of other auxiliary methods and technologies, such as chemotaxonomy and cytotaxonomy, is necessary to identify species within this genus [2]. Previous phytochemical investigations on the plants of this genus C. fissa, C. cuspidata var. seiboldii, and C. hystrix have led to the isolation of triterpene hexahydroxydiphenoyl (HHDP) esters, HHDP glucoses, galloyl, acylated quinic acids, phenol glucosides, condensed tannins, and flavonol glycosides [3–8].

In this study, we investigated C. fargesii, which is widely distributed in southern China, where it is usually used as a traditional medicine for the treatment of diarrhea, hemorrhage, and chronic ulcers [1]. Our preliminary analysis by HPLC and TLC indicated that the leaves are rich in tannins. Subsequent chromatographic separation of the extract yielded two metabolites 1 and 2, which were identified as 2,3-dihydroxy-2-(2-hydroxyethyl)-butanoic acid derivatives, and two phenolic compounds 3 and 4. This paper reports the isolation and structural characterization of the new compounds 1–3 and an assessment of the cytotoxicity of these molecules.

2. Results and Discussion

The fresh leaves of C. fargesii were extracted with 80% aqueous ethanol, and the extract was partitioned between Et2O and water. The Et2O and aqueous fractions were separated by a combination
of Sephadex LH-20, MCI gel CHP 20P, Toyopearl Butyl-650C, Chromatorex ODS, and Diaion HP20SS column chromatography and semi-preparative reverse-phase HPLC, to yield three new compounds 1–3 and one known phenolic compound gentisic acid 5-O-α-D-rhamnopyranosyl-(1→2)-β-D-glucoside [9] (4) (Figure 1).

![Figure 1. Structures of compounds 1–4.](image)

Compound 1 was isolated as a brown amorphous powder and gave a positive FeCl₃ test (dark blue), which suggested the presence of phenol moieties in the molecule. The molecular formula C₁₉H₂₃O₁₃ was determined based on the liquid chromatography-mass spectrometry IT-TOF (LC-MS/IT-TOF), which showed [M − H]⁻ and [M + Na]⁺ ion peaks at m/z 459.1143 (calcd. for C₁₉H₂₃O₁₃, 459.1144) and 483.1118 (calcd. for C₁₉H₂₄O₁₃Na, 483.1109), respectively. In the 1H- and 13C-NMR spectra (Table 1), two proton singlets at δH 4.32 and 4.58 suggested esterification with the galloyl moiety at this position. This was confirmed by the HMBC correlation between the hydroxyl proton and the carboxyl carbon (δC 175.4) and by the lower field shift of the carboxyl carbon (δC 175.4) compared to the ester carbonyl carbon (δC 173.3) in the remaining moiety. Consequently, the structure of compound 1 was established as 3-[1′-(6″-O-galloyl-β-D-glucopyranosyl)oxyethyl]-3-hydroxy-dihydrofuran-2(3H)-one.

Compound 2 was isolated as a yellow amorphous powder and gave a positive FeCl₃ test (dark blue). The presence of a galloyl group was deduced from the 13C-NMR signals (Table 1) [6]. The molecular formula of C₁₃H₁₆O₉ was established based on the LC-MS/IT-TOF (m/z 339.0672...
[M + Na]+, calcd. for C_{15}H_{16}O_{6}Na, 339.0687) and $^{13}$C-NMR data. The $^{13}$C-NMR and DEPT spectra showed signals attributable to a methyl ($\delta_C$ 16.9), a methylene ($\delta_C$ 34.2), an oxymethine ($\delta_C$ 71.4), an oxymethylene ($\delta_C$ 60.2), an oxy quaternary ($\delta_C$ 77.7), and a carboxyl ($\delta_C$ 175.4) carbons. The NMR data of 2 (Table 1) were similar to those of 1, except for the absence of the signals for one glucosyl moiety, which was supported by its MS data. A 2,3-dihydroxy-2-(2'-oxyethyl)-butanoic acid moiety could be constructed by $^1$H-$^1$H COSY correlations (Figure 2) of H-4 with H-3 and H-1 with the carboxy carbon ($\delta$C 60.2), an oxymethine ($\delta$C 71.4), H-3 with C-1, H-1 with C-2, and the HMBC correlations of H-4 with C-3 ($\delta_C$ 71.4), H-3 with C-1, H-1 with C-1, and H-2 with C-2 (Figure 2). The HMBC correlation of H-2' with the carboxy carbon ($\delta$C 165.9) indicated that the galloyl group was attached to C-2'. Based on these results, the structure of 2 was determined to be 2-[2'(galloyl)-oxyethyl]-2,3-dihydroxybutanoic acid.

**Table 1.** $^1$H (500 MHz) and $^{13}$C (125 MHz) NMR data of compounds 1 and 2 in acetone-$d_6$.

| Positions | 1 | 2 |
|-----------|---|---|
| $^1$H     | $^{13}$C | $^1$H | $^{13}$C |
| 1         | 178.6 | 175.4 |
| 2         | 76.6  | 77.7  |
| 3         | 2.10 (1H, m) | 3.88 (1H, q, $J$ = 6.5 Hz) | 29.9  |
|           | 2.58 (1H, m) | 6.4 Hz | 16.9  |
| 4         | 4.31-4.34 (2H, m) | 2.31 (1H, m), 2.36 (1H, m) | 104.1 |
| 1'        | 4.01 (1H, d, $J$ = 6.4 Hz) | 15.6  |
| 2'        | 1.26 (1H, d, $J$ = 6.4 Hz) | 4.33 (1H, m) | 60.2  |
| 1''       | 4.41 (1H, d, $J$ = 7.8 Hz) | 104.1 |
| 2''       | 3.19 (1H, d, $J$ = 7.8, 8.9 Hz) | 73.8  |
| 3''       | 3.43 (1H, t, $J$ = 8.9 Hz) | 76.4  |
| 4''       | 3.46 (1H, t, $J$ = 8.9 Hz) | 70.4  |
| 5''       | 3.62 (1H, m) | 73.9  |
| 6''       | 4.32 (1H, m) | 63.8  |

Galloyl  
| Positions | 1 | 2 |
|-----------|---|---|
| 1         | 7.13 (2H, s) | 7.10 (2H, s) |
| 2,6       | 109.1 | 109.1 |
| 3,5       | 145.2 | 145.0 |
| 4         | 138.1 | 137.7 |
| 7         | 166.4 | 165.9 |

**Figure 2.** Key HMBC and $^1$H-$^1$H COSY correlations of 1 and 2.

The absolute configurations at C-2 and C-3 of 2 were established using the modified Mosher’s method [11,12]. Treatment of 2 with CH$_3$I, then with (R)-(−) and (S)-(+)2-methoxy-2-trifluoromethyl-2-phenylacetyl (MTPA) chloride to get the C-3 (S)- and (R)-MTPA ester derivatives, respectively. Δδ values obtained from the $^1$H-NMR data of the C-3 (R)- and (S)-MTPA ester derivative indicated that the absolute configuration at C-3 of 2 was S (Figure 3). Compound 2 reacted with 2,2-dimethoxypropane (DMP) and pyridinium p-toluene sulfonate (PPTS) to form the 2,3-O-isopropylidene derivative. The C-2 and C-3 relative configuration of 2,3-O-isopropylidene derivative was determined based on the NOE correlation of H-1’ with H$_3$-4 (Figure 4). Thus, the absolute configurations at C-2 and C-3 of 2 were assigned as R and S, respectively.
Compounds 1 and 2 both contain a 2,3-dihydroxy-2-(2-hydroxyethyl)-butanoic acid moiety. The hydrolysis of 1 in 1 M HCl yielded 3-hydroxy-3-(1-hydroxyethyl)dihydrofuran-2(3H)-one that was identified to have the same absolute configuration as 2 by comparing their [α]D25 and CD data. Hence, the absolute configurations of 1 were assigned as 3R,1’S.

Compound 3 was obtained as a brown amorphous powder, which gave a dark blue color with FeCl3. The molecular formula C17H16O8 was deduced from the [M−H]− peak at m/z 347.0768 in the LC-MS/IT-TOF (calcd. for C17H15O8, 347.0772). Comparison of the 1H- and 13C-NMR data of 3 (Table 2) and (3S,4S)-3-[(β-D-glucopyranosyl)oxymethyl]-3,4-dihydro-5,6,7-trihydroxy-4-(4’-hydroxy-3’-methoxyphenyl)-1H-[2]benzopyran-1-one [13] revealed that the methyl at C-3 in the known compound was replaced by a hydroxymethyl in 3. This was confirmed by the MS data and the correlations of the methine proton (δH 4.67) with the methylene carbon (δC 62.4) in the HMBC spectrum (Figure 5), as well as 1H-1H COSY correlations of H-3 (δH 4.67) with H-3a (δH 3.54) and H-4 (δH 4.50) (Figure 5). Comparison of the CD and the optical rotation ([α]D25 +18.3°) data of 3 with those of similar compounds suggested that the absolute configuration is 3S,4S [14]. Based on the above evidences, the structure of compound 3 was concluded to be (3S,4S)-3-hydroxymethyl-3,4-dihydro-5,6,7-trihydroxy-4-(4’-hydroxy-3’-methoxyphenyl)-1H-[2]-benzopyran-1-one.

Table 2. 1H-NMR (500 MHz) and 13C-NMR (125 MHz) data of compound 3 in DMSO-d6.

| Positions | 1H            | 13C          |
|-----------|---------------|--------------|
| 1         | 164.6         |              |
| 2         |               |              |
| 3         | 4.67 (1H, td, j = 1.1, 6.5 Hz) | 84.7         |
| 3a        | 3.54 (1H, dd, j = 7.6, 11.2 Hz)  | 62.4         |
| 3b        | 3.68 (1H, dd, j = 6.5, 11.2 Hz)  |              |
| 4         | 4.50 (1H, br s) | 36.7         |
| 5         |               | 144.7        |
| 6         |               | 139.7        |
| 7         |               | 145.3        |
| 8         | 7.15 (1H, s)  | 107.7        |
| 9         |               | 115.3        |
| 10        |               | 119.9        |
| 1’        |               | 133.5        |
| 2’        | 6.83 (1H, d, j = 2.1 Hz)  | 111.8        |
| 3’        |               | 147.6        |
| 4’        |               | 144.8        |
| 5’        | 6.67 (1H, d, j = 8.1 Hz)  | 114.9        |
| 6’        | 6.64 (1H, dd, j = 2.1, 8.1 Hz) | 119.6        |
| OCH3      | 3.72 (3H, s)  | 55.4         |

Figure 3. Δδ(S-R) values of MTPA ester derivative of 2.

Figure 4. Key NOE correlations of 2,3-O-isopropylidene derivative of 2.
All isolates were subjected to a cytotoxicity assay in vitro against human lung epithelial A549, human hepatocellular carcinoma SMMC-7721 cell, human gastric carcinoma MGC-803 cell, liver hepatocellular HepG2 cell, and human breast adenocarcinoma MCF-7 tumour cell. Unfortunately, none of the isolates showed inhibitions of those tumour cells at the highest concentration tested (IC₅₀ value > 10 μM).

3. Experimental Section

3.1. Materials

The leaves of C. fargesii were collected at Guangxi Institute of Botany, Guangxi, China, in August 2014, and were identified by Prof. Shi-Hong Lu. A voucher specimen (20140627) was deposited in the Guangxi Key Laboratory of Functional Phytochemicals Research and Utilization, Guangxi Institute of Botany, China.

3.2. General Experimental Procedures

Optical rotations were measured with a 341 digital polarimeter (Perkin-Elmer Corp., Waltham, MA, USA). ¹H- and ¹³C-NMR spectra were measured in acetone at 27 °C, using an Avance 500 spectrometer (500 MHz for ¹H and 125 MHz for ¹³C, Bruker Biospin AG, Fällanden, Switzerland). Coupling constants and chemical shifts were given in Hz and on a δ (ppm) scale, respectively. GC was performed on a 6890N instrument equipped with a FID detector (Agilent Technologies, Santa Clara, CA, USA) operated at 280 °C (column: 28 m × 0.32 mm i.d. HP-5, column temp. 160 °C). LC-MS/IT-TOF was recorded on a LCMS-IT-TOF spectrometer (Shimadzu, Kyoto, Japan). Semi-preparative HPLC was performed on an Agilent 1200 apparatus equipped with a UV detector and a Zorbax SB-C-18 (9.4 × 250 mm) column (Agilent). Column chromatography (CC) was performed using Sephadex LH-20 (25–100 μm; GE Healthcare Bio-Science AB, Uppsala, Sweden), MCI gel CHP 20P (75–150 μm; Mitsubishi Chemical, Tokyo, Japan), Diaion HP20SS (Mitsubishi Chemical), Chromatorex ODS (100–200 mesh; Fuji Silysia Chemical, Aichi, Japan), and Toyopearl Butyl-650C (TOSOH, Tokyo, Japan) columns. TLC was performed on precoated Kieselgel 60 F₂₅₄ plates (0.2 mm thick; Merck, Darmstadt, Germany) with toluene–HCO₂Et–HCO₂H (1:7:1, v/v) as the solvent, and spots were detected by spraying with a 2% ethanolic FeCl₃.

3.3. Extraction and Separation

The fresh leaves of C. fargesii (5.20 kg) were cut into small pieces and extracted three times with EtOH/H₂O (8:2, v/v, 36 L) by maceration at room temperature for 7 days. The extracts were combined and concentrated under reduced pressure to give an aqueous solution. The solution was partitioned with Et₂O four times to give the Et₂O fraction (32.4 g). The aqueous layer was subjected to Sephadex LH-20 CC (8 cm i.d. × 40 cm) with 0%–100% MeOH–H₂O (20% stepwise elution, each 1.5 L) to give 9 fractions: frs 1 (15.6 g), 2 (84.5 g), 3 (26.6 g), 4 (27.0 g), 5 (130.0 g), 6 (12.9 g), 7 (3.3 g), 8 (2.3 g), 9 fractions: frs 1 (15.6 g), 2 (84.5 g), 3 (26.6 g), 4 (27.0 g), 5 (130.0 g), 6 (12.9 g), 7 (3.3 g), 8 (2.3 g),
and 9 (2.2 g). Fraction fr. 2 (84.5 g) was separated by MCI gel CHP 20PCC (6 cm i.d. × 40 cm) with MeOH–H2O (10% stepwise elution, each 1.0 L) to yield seven fractions, and fraction fr. 2-2 (5.3 g) was further fractionated by Diaion HP20SS CC (4 cm i.d. × 30 cm) with H2O containing increasing proportions of MeOH (0%–100%, 10% stepwise elution, each 0.5 L) to give 4 (105 mg). The Et2O fraction was subjected to MCI gel CHP 20PCC (5 cm i.d. × 50 cm) with 0%–100% MeOH in H2O (10% stepwise elution, each 0.5 L) to yield 10 fractions: frs E-1 (5.3 g), 2 (7.5 g), 3 (1.6 g), 4 (3.2 g), 5 (4.3 g), 6 (1.5 g), 7 (2.3 g), 8 (1.0 g), 9 (9.9 g) and 10 (3.5 g). Fr. E-3 was fractionated by Toyopearl Butyl-650C CC (3 cm i.d. × 30 cm) with 0%–100% MeOH–H2O containing 0.1% CF3CO2H (TFA) (10% stepwise elution, each 0.3 L) to give fr. E-31 (1.3 g) and fr. E-32 (122 mg). Fr. E-32 was further purified by Chromatorex ODS CC (3 cm i.d. × 30 cm) with 0%–100% MeOH in H2O (5% stepwise elution, each 0.2 L) to give 3 (12 mg). Fraction E-4 was separated by Sephadex LH-20 CC (4 cm i.d. × 40 cm) with H2O containing increasing amounts of MeOH (0%–100%, 10% stepwise elution, each 0.5 L) to yield Fr. E-41 (250 mg), Fr. E-42 (150 mg), Fr. E-43 (296 mg) and Fr. E-44 (1.7 g). The Fr. E-42 and Fr. E-43 were further purified by semi-preparative HPLC (MeCN/H2O, 20:80, 2.5 mL/min) to give 1 (46 mg, tR 14.5 min) and 2 (68 mg, tR 13.2 min), respectively.

3.4. Spectroscopic Data

\[ \text{(3S,1'R'S)-1'-(6"-O-Galloyl-β-D-glucopyranosyl)oxyethyl]-3-hydroxy-dihydrofuran-2(3H)-one (1):} \]
Brown amorphous powder; [α]D 25° +52.1° (c = 0.12, MeOH); UV (MeOH) \( \lambda_{\text{max}} \) nm (log ε): 272 (4.32); CD (MeOH) \( \lambda_{\text{max}} \) (Δε) 278 (8.4), 254 (4.5), 209 (2.7). 1H- and 13C-NMR data, see Table 1; LC-MS/IT-TOF m/z [M + H]⁺ 459.1143 (calcd. for C19H25O13, 459.1144) and [M + Na⁺] 483.1118 (calcd. for C19H25O13Na, 483.1109).

\[ \text{(3R,1'S)-3-Hydroxy-1'-hydroxyethyl-dihydrofuran-2(3H)-one (Hydrochloride of 1):} \]
[α]D 25° −17.0° (c = 0.15, MeOH); UV (MeOH) \( \lambda_{\text{max}} \) nm (log ε): 265 (3.16); CD (MeOH) \( \lambda_{\text{max}} \) (Δε) 267 (5.7), 251 (3.2), 211 (1.2). 1H-NMR (MeOH-d₄, 500 MHz) δ 4.36 (1H, m, H-5a), 4.25 (1H, m, H-5b), 2.11 (1H, m, H-4a), 2.33 (1H, m, H-4b), 1.21 (3H, d, J = 6.5 Hz, H-1'), 1.21 (3H, d, J = 6.5 Hz, H-2'); LC-MS/IT-TOF m/z 169.0475 [M + Na⁺] (calcd. for C₆H₁₀O₄Na, 169.0477).

\[ \text{(2R,3S)-2-(12"-Galloyloxoyethyl)-4-hydroxybutanoic acid (2):} \]
Yellow amorphous powder; [α]D 25° −16.7° (c = 0.12, MeOH); UV (MeOH) \( \lambda_{\text{max}} \) nm (log ε): 267 (4.26); CD (MeOH) \( \lambda_{\text{max}} \) (Δε) 268 (10.6), 252 (4.6), 212 (1.4). 1H- and 13C-NMR data, see Table 1; LC-MS/IT-TOF m/z 339.0672 [M + Na⁺] (calcd. for C₁₃H₁₄O₆Na, 339.0687).

\[ \text{(3S,4S)-3-Hydroxymethyl-3,4-dihydro-5,6,7-trihydroxy-4-(4'-hydroxy-3'-methoxyphenyl)-1H-2]-benzopyran-1-one (3):} \]
Brown amorphous powder; [α]D 25° +18.3° (c = 0.11, MeOH); UV (MeOH) \( \lambda_{\text{max}} \) nm (log ε): 220 (4.26), 278 (2.35); CD (MeOH) \( \lambda_{\text{max}} \) (Δε) 288 (11.4), 242 (6.5), 218 (2.7). 1H- and 13C-NMR data, see Table 2; LC-MS/IT-TOF m/z 347.0768 [M − H]⁻ (calcd. for C₁₇H₁₅O₈, 347.0772).

3.5. Preparation of MTPA Ester Derivatives

CH₃I (30 mg) and K₂CO₃ (15 mg) were added to a solution of 2 (10 mg) in DMF (5 mL). After stirring for 24 h at room temperature (r.t.), the reaction mixture was suspended in H₂O and extracted with CHCl₃. The CHCl₃ layer was vacuum dried to afford a residue (6.2 mg). Then, DMAP (3.8 mg), Et₃N (4.0 μL), and (R)-(−)-MTPACl (3.0 μL) were added to a solution of the residue (3.1 mg) in CH₂Cl₂ (1.0 mL) and stirred for 4 h at r.t. The reaction mixture was dried under a stream of N₂. Separation of the residue was done by a silica gel column (hexane/EtOAc, 4:1) to afford the (S)-MTPA ester derivative (2.1 mg). The (R)-MTPA ester derivative (2.3 mg) was obtained according to the same procedure using (S)-(−) MTPACl.

(3S)-MTPA Ester derivative of 2: Colorless oil; 1H-NMR (MeOH-d₄, 500 MHz) δ 7.0512–7.3901 (7H), 4.8212 (1H, q, J = 6.6 Hz, H-3), 1.3648 (3H, d, J = 6.6 Hz, H-4), 2.2511 (1H, m, H-1’a), 2.3004 (1H, m,
H-1′b), 4.2526 (1H, m, H-2′a), 4.2812 (1H, m, H-2′b), 3.3206–3.8516 (-OCH3 × 5). LC-MS/IT-TOF m/z 611.17147 [M + Na]+ (calcd. for C27H31F3O11Na, 611.17162).

(3R)-MTPA Ester derivative of 2: Colorless oil; 1H-NMR (MeOH-d4, 500 MHz) δ 7.0510–7.3902 (7H), 4.8210 (1H, q, J = 6.6 Hz, H-3), 1.3608 (3H, d, J = 6.6 Hz, H-4), 2.2572 (1H, m, H-1′a), 2.3088 (1H, m, H-1′b), 4.2548 (1H, m, H-2′a), 4.2858 (1H, m, H-2′b), 3.3312–3.8716 (-OCH3 × 5). LC-MS/IT-TOF m/z 611.1710 [M + Na]+ (calcd. for C27H31F3O11Na, 611.1716).

3.6. Preparation of Acetonide Derivative of 2

Compound 2 (10.2 mg) was dissolved in acetonitrile (1.0 mL) and treated with DMAP (0.2 mL) and PPTS (6.5 mg) at r.t. After 4 h, Et3N (7.5 µL) was added and the mixture was concentrated by N2 blowing. The residue was separated on a silica gel column (CH2Cl2/MeOH, 4:1–2:1) to afford the 2,3-O-isopropylidene derivative (3.2 mg) of 2.

2,3-O-Isopropylidene Derivative of 2: Colorless oil; 1H-NMR (MeOH-d4, 500 MHz) δ 6.98 (2H, s), 4.36 (1H, q, J = 6.5 Hz, H-3), 1.21 (3H, d, J = 6.5 Hz, H-4), 2.11 (1H, m, H-1′a), 2.23 (1H, m, H-1′b), 4.25 (1H, m, H-2′a), 4.29 (1H, m, H-2′b), 1.27 (3H, s, acetone-CH3), 1.31 (3H, s, acetone-CH3); LC-MS/IT-TOF m/z 379.1007 [M + Na]+ (calcd. for C18H20O9Na, 379.1005).

3.7. Acid Hydrolysis and Sugar Analysis by GC

Compound 1 (6 mg) was dissolved in MeOH (4.0 mL) and 1 M H2SO4 (2.0 mL) and refluxed for 2 h on a H2O bath. After the hydrolysate was cool, H2O (8.0 mL) was added, then extracted with EtOAc (3 × 10.0 mL). The EtOAc layer was vacuum dried and chromatographed on semi-preparative HPLC eluting with a gradient of MeOH–H2O (5:95–25:75, v/v) to afford 3-hydroxy-1′,2′-hydroxyethyl-dihydrofuran-2(3H)-one. The aqueous layer was neutralized with aqueous Ba(OH)2 and evaporated under reduced pressure to give a residue. The residue was dissolved in pyridine (100 µL), subsequent treated with 0.1 M cysteine methyl ester hydrochloride (150 µL; Sigma, St. Louis, MO, USA) and warmed at 60 °C for 1 h, then the trimethylsilylation reagent HMDS/TMCS (hexamethyldisilazane/trimethylchlorosilane/pyridine 2:1:10; Acros Organics, Geel, Belgium) was added and warmed at 60 °C for 30 min. The reaction mixture was partitioned between water and hexane. The hexane extract was analyzed by GC [15] (detector temperature: 280 °C; injector temperature: 250 °C; temperature gradient: start at 160 °C, hold for 5 min, increase to 280 °C at 5 °C/min, hold for 10 min). The authentic samples were analyzed in the same way. The tR values of D-glucose and L-glucose were 13.25 min and 15.32 min, respectively. The thiazolidine derivatives of the samples were confirmed by comparison with authentic standards.

3.8. Cytotoxicity Assay

All isolates were tested for cytotoxicity in vitro against A549, SMMC-7721, MGC-803, HepG2, and MCF-7 tumour cells via the MTT assay [16,17] with hydroxycamptothecine as a positive control.

4. Conclusions

In this study, we separated and identified three new compounds 1–3 and a known compound 4 from the leaves of C. fargesii. The 2,3-dihydroxy-2-(2-hydroxyethyl)-butanoic acid moiety in 1 and 2 is an unusual carboxylic acid in Nature, thus, these compounds might be recognized as chemotaxonomic markers. Our preliminary examination also suggested this plant contains triterpene HHDP esters, which are important chemotaxonomical markers of Castanopsis sp.; therefore, further phytochemical investigations of the leaves of C. fargesii are in progress.

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Sample Availability: Samples of the compounds 1–4 are available from the authors.