ABSTRACT: Five metals in sediment samples at seven sites from the Ambon Bay were analyzed with BCR sequential extraction procedure to determine chemical fractionation of metals and to assess bioavailability of metals with Risk Assessment Code (RAC). The result showed that the percentages of cadmium (100%), lead (82.6-97.08%) and zinc (41.68-76.33%) were mostly accumulated in the non-residual (F1+F2+F3) fraction of the total concentrations. While the copper percentages (44.74-78.91%) and nickel (59.71-74.16%) were mostly accumulated in residual (F4) fraction of the total concentrations. The Risk Assessment Code (RAC) reveals that cadmium, copper, nickel and zinc at locations exist in acid soluble (exchangeable) fraction and therefore, they are in low until very high risk category meanwhile there is no Pb at locations exist in acid soluble (exchangeable) fraction.

Keywords: metal fractionation; sediment; RAC; Ambon Bay.

INTRODUCTION

Pollution of heavy metal is a critical environmental issue because of the nature of the contaminants is persistent and not biodegradable (Sarkar et al., 2014). Sediments are the ultimate sink for heavy metals of both natural and anthropogenic origin (Campbel et al., 1988), also for pollution (Hlavay et al., 2004). Toxicity and mobility of metals are strictly related to their chemical form; for this reason, the knowledge of the total metal content of a matrix does not allow to achieve information on their bioavailability and eco-toxicology. Trace metals in sediments and soils can be associated with several geochemical phases, the most important of which can identify as carbonates, iron and manganese oxides, sulfides, organic matter. Sequential extraction procedures allow the individuation of the metals partitioning along the fractions of solid environmental matrices, providing therefore, useful information on their bioavailability and mobility (Craba et al., 2004).

Ambon City is the capital of Moluccas Province that surroundings by Ambon Bay. It is separated by a shallow sill (≤12m depth) between the outer and inner part as geographically (Corvianawatie et al., 2016). Ambon Bay is the coastal area that is vulnerable to pollution due to anthropogenic activities such as urban, farming, shipping activities, and ship repair activities from the dockyard around the Ambon Bay that may contribute to metal inputs to the Ambon Bay (Manullang et al., 2017). In Ambon Bay, total metal concentration in sediment has been reported by many studies (Siahaya et al., 2013, Tupan et al., 2013, Manullang et al., 2017).
Metal bioavailability in sedimentary environmental samples may provide a better understanding of the interaction of environment-organisms. The bioactivity and bioavailability of metals is very dependent on the forms of chemical connections, and their speciation. Determination of total metal concentration in environmental substances is not enough to assess the environmental impact of polluted sediments because heavy metals may have different chemical forms and only a small portion can be easily remobilized. Studies on the distribution and speciation of heavy metals in sediments not only provide the information about pollution levels, but also especially actual environmental impacts on the bioavailability of metals and their origin (Benson et al., 2013).

The information on heavy metals speciation studies with different methods has been done in some in polluted areas in Indonesia such as Semarang City (Takarina et al., 2004), Dumai, Sumatera (Amin et al., 2007), Delta Berau East Kalimantan (Arifin et al., 2010), Jenebarang River (Najamuddin et al., 2016). The Community Bureau of Reference (BCR) protocols for sediment and soil give a reasonable basis for most of the solid samples, and the results can be compared among different laboratories (Hlavay et al., 2004). BCR method used in several studies such as Spermonde Island, Makassar (Werolangi et al., 2016) and Banten Bay (Lestari et al., 2018).

The aim of this study to determine the chemical fractionation of metals and to assess the bioavailability of metals with Risk Assessment Code. The result from this paper will provide a better understanding of environmental risk of metals in the surface sediment of the Ambon Bay.

METHOD

Sampling Procedure

Seven sediment samples (0 to 10cm from surface) at 10 to 12m water depth were collected using a stainless-steel grab from 25 to 28 August 2015, located in Ambon Bay. These sampling sites were in the area that is influenced by anthropogenic stress and water circulation. Three sampling areas located in Ambon Inner Bay (AIB), one site in Threshold Galala and the other three in Ambon Outer Bay (AOB), as featured in Figure 1 (Manullang et al., 2017).

Sediment samples that have been collected, were then labelled and stored in a cooler box before being taken to the laboratory. In laboratory, the sediment samples were dried at a temperature of 60°C until it reaches a constant weight, then homogenized using mortar. In the laboratory facility of the Research Centre for Oceanography, Indonesia Institute of Sciences, sediment samples were prepared for fractionation of metal analysis.

Fractionation of metals analysis

The fractionation method used a BCR three-step sequential extraction procedure from Cuong and Obbard (2006) with a small modification in the amount
of sample and reagent for residual step analysis. This method provides information on four fractions, namely, exchangeable and carbonates (F1); reducible (F2); oxidizable (F3); and residual (F4). The total content is determined by the sum of all fraction (F1+F2+F3+F4). The fractionation method is described in Figure 2.

Risk Assessment Code (RAC)

The result of this fractionation study showed that the metals in sediments are bound to different fractions with different strengths. These strength values give a clear indication of sediment reactivity supporting the risk assessment due to the metal existence in Ambon Bay. Many previous studies have used the criteria of Risk Assessment Code (RAC) to evaluate sediment quality (e.g. Jain, 2004; Turki, 2007; Horvath et al., 2013; Morelli and Gasparon, 2014). The criteria of Risk Assessment Code (RAC) indicated that sediment releasing exchangeable and carbonate fractions (F1) less than 1% of the total metal, was considered safe for the environment. On the contrary, the sediment releasing more than 50% of the total metal in the same fraction has been considered highly dangerous and high possibility to enter the food chain (Jain, 2004).

RESULT AND DISCUSSION

Total Metals Concentration in Sediment

The concentration of Cd, Cu, Ni, Pb and Zn in total and each geochemical fraction of coastal sediment from Ambon Bay are presented in Table 1. The sum of F1+F2+F3+F4 described total Cd, Cu, Ni and Pb in each station. The range concentrations of the five metals varied as follows; Cd (0.07-0.19 mg/kg dw), Cu (11.0-41.1 mg/kg dw), Ni (18.6-197 mg/kg dw), Pb (11.5-17.4 mg/kg dw) and Zn (35.3-147 mg/kg dw). The highest concentration of these metals was varied in these location. Total Cd was found in Passo (Station 1), meanwhile total Cu was found in Waiheru (Station 2). Total Ni was found in Poka (Station 3), Ambang Galala (Station 4) and Passo (Station 1), total Pb was found in Waihaong (Station 6) and Waiheru (Station 2) while total Zn was found in Ambang Galala (Station 4). The population in Waiheru and Poka are less dense than that of Passo. Meanwhile Batu Merah and Waihaong are area with a large population. The biggest port of Ambon is located around Waihaong, whereas the highest total Pb was detected. Concentration of Cd and Pb in this area were lower than in East China Sea (Andreas & Zhang, 2016). Cu concentration in this location is almost in the same amount as in Cirebon waters (Lestari, 2018), but lower than Muara Angke in Jakarta Bay (Lestari and Budiyanto, 2019), as the Zn concentration status.

Fractionation of Metal in sediment

The potential environmental risk of trace elements in sediments is associated with their total content either their fractionation. The chemical partitioning of the considered metals (Cd, Cu, Ni, Pb and Zn) from each extraction step has been described in Figure 3 and showed that the metals in each location were fractioned differently.
| No | Location          | F1 Acid Soluble | F2 Reducible | F3 Oxidizable | F4 Residual | ∑Cd Fraction |
|----|-------------------|----------------|--------------|---------------|-------------|--------------|
| 1  | Passo             | 0.14           | 0.02         | 0.03          | nd          | 0.19         |
| 2  | Waiheru           | 0.02           | 0.02         | 0.12          | nd          | 0.17         |
| 3  | Poka              | 0.02           | 0.02         | 0.12          | nd          | 0.17         |
| 4  | Ambang Galala     | 0.10           | 0.02         | 0.03          | nd          | 0.15         |
| 5  | Batu Merah        | 0.01           | 0.06         | 0.07          | nd          | 0.15         |
| 6  | Waihaong          | 0.02           | 0.02         | 0.03          | nd          | 0.07         |
| 7  | Tawiri            | 0.01           | 0.02         | 0.03          | nd          | 0.07         |

| No | Location          | F1 Acid Soluble | F2 Reducible | F3 Oxidizable | F4 Residual | ∑Cu Fraction |
|----|-------------------|----------------|--------------|---------------|-------------|--------------|
| 1  | Passo             | 0.47           | 1.85         | 1.25          | 13.3        | 16.9         |
| 2  | Waiheru           | 2.75           | 7.51         | 4.11          | 26.7        | 41.1         |
| 3  | Poka              | 1.45           | 3.95         | 1.27          | 11.6        | 18.2         |
| 4  | Ambang Galala     | 2.74           | 6.72         | 5.57          | 12.2        | 27.2         |
| 5  | Batu Merah        | 0.36           | 0.79         | 2.58          | 7.24        | 11.0         |
| 6  | Waihaong          | 0.26           | 1.88         | 2.89          | 16.3        | 21.3         |
| 7  | Tawiri            | 0.37           | 1.23         | 5.16          | 9.59        | 16.3         |

| No | Location          | F1 Acid Soluble | F2 Reducible | F3 Oxidizable | F4 Residual | ∑Ni Fraction |
|----|-------------------|----------------|--------------|---------------|-------------|--------------|
| 1  | Passo             | 9.21           | 14.4         | 14.1          | 69.1        | 107          |
| 2  | Waiheru           | 1.79           | 3.08         | 3.85          | 25.1        | 33.8         |
| 3  | Poka              | 15.4           | 26.2         | 31.7          | 124         | 197.0        |
| 4  | Ambang Galala     | 14.5           | 13.0         | 17.1          | 77.5        | 122.1        |
| 5  | Batu Merah        | 3.67           | 1.76         | 1.61          | 11.5        | 18.6         |
| 6  | Waihaong          | 3.24           | 2.43         | 2.44          | 21.8        | 29.9         |
| 7  | Tawiri            | 6.43           | 4.02         | 3.63          | 20.9        | 34.9         |

| No | Location          | F1 Acid Soluble | F2 Reducible | F3 Oxidizable | F4 Residual | ∑Pb Fraction |
|----|-------------------|----------------|--------------|---------------|-------------|--------------|
| 1  | Passo             | nd             | 9.25         | 1.33          | 2.23        | 12.8         |
| 2  | Waiheru           | nd             | 13.5         | 1.73          | 1.89        | 17.1         |
| 3  | Poka              | nd             | 9.44         | 2.11          | 0.35        | 11.9         |
| 4  | Ambang Galala     | nd             | 13.4         | 0.97          | 2.26        | 16.7         |
| 5  | Batu Merah        | nd             | 10.6         | 0.59          | 0.34        | 11.5         |
| 6  | Waihaong          | nd             | 14.2         | 0.60          | 2.66        | 17.4         |
| 7  | Tawiri            | nd             | 12.4         | 3.21          | 1.11        | 16.7         |

| No | Location          | F1 Acid Soluble | F2 Reducible | F3 Oxidizable | F4 Residual | ∑Zn Fraction |
|----|-------------------|----------------|--------------|---------------|-------------|--------------|
| 1  | Passo             | 3.01           | 17.3         | 4.37          | 27.5        | 52.2         |
| 2  | Waiheru           | 11.1           | 15.7         | 14.8          | 58.1        | 99.7         |
| 3  | Poka              | 36.8           | 18.0         | 5.87          | 36.8        | 97.5         |
| 4  | Ambang Galala     | 67.9           | 30.1         | 14.4          | 34.9        | 147          |
| 5  | Batu Merah        | 9.78           | 9.06         | 2.13          | 14.3        | 35.3         |
| 6  | Waihaong          | 13.9           | 11.3         | 4.93          | 40.2        | 70.3         |
| 7  | Tawiri            | 5.64           | 26.9         | 9.71          | 26.2        | 68.5         |
Cd concentrations in sediments are a major concern because of their high toxicity. BCR procedures have been used to obtain information about the distribution of Cd in sediments. The acid soluble fraction (F1) contributed a higher portion for Cd 33% in this study as well as several studies in other areas (Andreas & Zhang, 2016, Nemati et al., 2011, Qiao et al., 2013, Lin et. al., 2014, Yang et al., 2014, Cao et al., 2015, Lin et al., 2016, Gu, 2018). Cadmium in this fraction is easy to move, unstable, bioavailability, toxicity, and is therefore easily extracted from sediment by biota (Andreas & Zhang, 2016). Contribution of Cu only 5%, Ni 12%, Zn 22%, and there is no Pb in this fraction (F1). The F2 fraction is portion of a metal that is enclosed in Fe/Mn oxide or hydroxide precipitation, that is hard to move in consequence of hard ionic bonds. However, this can be deoxidized and lead to second pollution, if potential redox and oxygen in seawater declined (Lin et al., 2014). The reducible fraction (F2) contributed the highest portion to Pb (80%), meanwhile contribution to Cd only about 23%, Cu 14%, Ni 11%, and Zn 24%. In the F3 fraction, metal ions as the center of action are bound with reactive groups of organic materials or produce materials that are not dissolved in water with sulfur ions, as a result, it is complicated to be liberated under normal moderate reduction or weak oxidation environment (Lin et al., 2014). The organic fraction (F3) contributed highest portion for Cd 45%, while for Cu only about (16%), Ni (12 %), Pb (10%), and Zn (9%). In the F4 fraction, metal ions are bound to the aluminosilicate lattice, due to their strong stability, lowest bioactivity and toxicity so they are considered to have no bioavailability (Lin et al., 2014). The residual fraction (F4) is the most important fraction for Cu (65%), Ni (66%) and Zn (44%) whilst for Pb only 10% and no Cd contribution. Thus, Cd fractionation mainly consists of F1 and F3 fractions, Pb bound to F2, Cu, Ni, and Zn are dominated by F4 in surface sediments in Ambon Bay. The result of this study is almost similar to the condition in Xiamen Bay (Lin et al., 2014). In Muara Angke, Jakarta Bay, the main content of Zn is F1, Pb is strongly tied to F2, Cu is bound to F3 while Ni dominates F4 (Lestari and Budiyanto, 2019). While in the waters of Cirebon and Banten Bay, Cu, Ni and Zn are dominated by F4 (Lestari et al. 2018).

Figure 3 showed that the distribution of the heavy metals in percent bar graphs in the four fractions (F1 is acid soluble fraction, F2 is reducible fraction, F3 is oxidizable fraction, and F4 is residual fraction) obtained in our analysis. The partitioning showed of metals associated with the non-residual fraction (consists of F1 + F2 + F3) and the residual fraction. The partitioning showed that the percentages of 3 metals (Cd, Pb and Zn) are more associated to the non-residual fractions (100% for Cd; 90% for Pb and 55.9% for Zn) than to the residual fraction. This indicates that these metals are derived from anthropogenic inputs instead of geochemical background (especially for Cd and Pb). In contrast, two metals (Cu and Ni) are greater in residual fraction (F4) with percentage 64.7% and 65.7% respectively. Therefore, the results indicated that the metals are potentially more available for exchange and/or release into the marine environment. The studied metals in the non-residual fractions increased in the order of Cd>Pb>Zn>Cu> Ni.

The proportion of cadmium 100%, lead 82.6-97.08% and zinc 41.68-76.33% were mostly accumulated in the non-residual fraction of the total concentrations, while some parts of copper 44.74-78.91% and nickel 59.71-74.16% were mostly accumulated in residual fraction of the total concentrations (Figure 3).

**Risk Assessment Code**

The Risk Assessment Code (RAC) reveals that about 10-72.73% of cadmium, 1.2-10.1% of copper, 5.30-19.75% of nickel, 8.24-46.12% of zinc and <1 of lead at the location exist in the acid soluble fraction or exchangeable fraction. Therefore, Cd can be categorized as medium risk – very high risk, Zn is in low risk – high risk category, nickel and copper are considered low – medium risk and no risk for lead (Table 2). Therefore, monitoring of the presence of metal in the waters is need to be done because it is persistent, it can be accumulated, and it endanger the aquatic environment.

Table 2. Risk Assessment Code (after Jain, 2004).

| Risk Assessment Code (RAC) | Criteria (%) |
|---------------------------|--------------|
| No Risk                   | < 1          |
| Low Risk                  | 1—10         |
| Medium Risk               | 11—30        |
| High Risk                 | 31—50        |
| Very High Risk            | > 50         |

**CONCLUSIONS**

Sediments from the coast of Ambon Bay have been analyzed using sequential extraction procedures BCR. Speciation studies of Cd, Cu, Pb, Ni, and Zn in Ambon Bay sediments showed that Cd fractionation mainly consists of soluble acid fraction (F1) and oxidized fraction (F3), Pb is bound to the reducible fraction (F2), whereas Cu, Ni and Zn are dominated by residual fraction (F4), however contribution of total non-residual fractions (F1+F2+F3) to element Zn is actually still higher compared to the contribution of F4.
Therefore it can be concluded that cadmium, lead and zinc are mostly accumulated in non-residual fractions meanwhile Cu and Ni were mostly accumulated in residual fraction of total concentration. The studied metals in the non-residual fractions increased in the order of Cd>Pb>Zn>Cu> Ni indicating they are originated from anthropogenic inputs rather than geochemical backgrounds. The Risk Assessment Code (RAC) reveals that cadmium, copper, nickel and zinc in these area are in fractions that are soluble in acids (exchangeable). In general they are in the low to very high risk category for the environment. There is no Pb in the acid soluble fraction (exchangeable) in the area.

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REFERENCES

Amin, B., Ismail, A., Arshad, A., and Kamarudin, M.S. 2007. Distribution and speciation of heavy metals (Cd, Cu, and Ni) in coastal sediment of Dumai Sumatera, Indonesia. Journal of Coastal Development, 10(2): 125 – 141.
Andreas, R., and Jing, Z. 2016. Fractionation and environmental risk of trace metals in surface sediment of The East China Sea by modified BCR sequential extraction method. *Molekul*, 11(1): 42 – 52.

Arifin, Z., Situmorang, S.B., and Booij, K. 2010. Geochemistry of heavy metals (Pb, Cr and Cu) in sediments and benthic communities of Berau Delta, Indonesia. *Coastal Marine Science*, 34(1): 205-211.

Benson, N.U., Anake, W.U., and Olanrewaju, I.O. 2013. Analytical relevance of trace metal speciation in environmental and biophysical system. *American Journal of Analytical Chemistry*, 4: 633-641.

Campbell, P.G.C., Lewis, A.G., Chapman, P.M., Crowder, A.A., Fletcher, W.K., Imber, B., Luoma, S.N., Stokes, P.M., and Winfrey, M. 1988. Biologically available metals in sediments, *National Research Council of Canada, Publication No. NRCC 27694*.

Cao L., Tian, H., Yang, J., Shi, P., Lou, Q., Waxi, L., Ni, Z., and Peng, X. 2015. Multivariate analyses and evaluation of heavy metals by Chemometric BCR sequential extraction method in surface sediments from Lingdingyang Bay, South China. *Sustainability*, 7: 4938-4951; doi:10.3390/su7054938.

Corvianawatie, C., Cahyarini, S.Y . and Putri, M.R. 2016. Reconstruction of sea surface temperature data based on the Sr/Ca of *Porites* Coral in Ambon Bay. *Journal of Mathematical and Fundamental Sciences*, 48(2): 115-129.

Craba, L., Brunori, C., Galletti, M., Cremisini, C., and Morabito, R. 2004. Comparison of three sequential extraction procedures (original and modified 3 steps BCR procedure) Applied to sediments of different origin. *Annali Di Chimica*, 94(56): 409–419. doi:10.1002/adic.200490050.

Cuong, D.T., and Obbard, N. 2006. Metal speciation in coastal marine sediments from Singapore using a modified BCR-sequential extraction procedure. *Applied Geochemistry*, 21: 1335–1346. doi: 10.1016/j.apgeochem.2006.05.001.

Gu, Y-G. 2018. Heavy metal fractionation and ecological risk implications in the intertidal surface sediments of Zhelin Bay, South China. *Marine Pollution Bulletin*, 129(2): 905-912.

Hlavay, J., Prohaska, T., Weisz, M., Wenzel, W.W., and Stingeder, J. 2004. Determination of trace elements bound to soils and sediment fractions (IUPAC Technical Report). *Pure Applied Chemistry*, 76(2): 415–442.

Horvath, M., Halasz, G., Kucanova, E., Kucikova, B., Fekete, I., Remetieova, D., Heltai, G., and Florian, K. 2013. Sequential extraction studies on aquatic sediment and biofilm samples for the assessment of heavy metal mobility. *Microchemical Journal*, 107: 121-125.

Jain, C. K. 2004 Metal fractionation study on bed sediments of River Yamuna, India. *Water Research*, 38; 569-578.

Lestari and Budiyanto F. 2019. Metal speciation in sediment from Muara Angke, Jakarta Bay using of BCR sequential extraction procedure, *Ilmu Kelautan: Indonesian Journal of Marine Sciences*, 24 (1): 23-30.

Lestari, Budiyanto F. and Hindarti D. 2018. Speciation of heavy metals Cu, Ni and Zn by modified BCR sequential extraction procedure in sediments from Banten Bay, Banten Province, Indonesia. *Global Colloquium on GeoSciences and Engineering 2017, IOP Conf. Series: Earth and Environmental Science 118 (2018) 012059* doi :10.1088/1755-1315/118/1/012059.

Lin, C., Liu Y ., Li W., Sun X., and Ji W. 2014. Speciation, distribution, and potential ecological risk assessment of heavy metals in Xiamen Bay surface sediment. *Acta Oceanolica Sinica*, 33 (4): 13–2. DOI: 10.1007/s13131-014-0453-2.

Lin Y ., Meng F., Du Y. and Tan Y. 2016. Distribution, speciation, and ecological risk assessment of heavy metals in surface sediments of Jiaozhou Bay, China. *Human and Ecological Risk Assessment: An International Journal*, 22(5): 1253-1267, DOI: 10.1080/10807039.2016.1159503.

Manullang C.Y ., Lestari, Tapilatu Y .and Arifin Z. 2017. Assessment of Fe, Cu, Zn, Pb, Cd & Hg in Ambon Bay surface sediments. Marine. Research in. *Indonesia Journal*, 42 (20): 77-86.

Morelli G and Gasparon M., 2014. Metal contamination of estuarine intertidal sediments of Moreton Bay, Australia. *Marine Pollution Bulletin*, 89: 435-443 .

Najamuddin, Prartono T, Sanusi H.S. and Nurjaya I.W. 2016. Seasonal distribution and geochemical fractionation of heavy metals from surface sediment in a tropical estuary of Jenebarang River, Indonesia. *Marine Pollution Bulletin*, 111: 456-462.
Nemati, K., Abu Bakar, N. K., Abas, M. R., and Sobhanzadeh, E. 2011. Speciation of heavy metals by modified BCR sequential extraction procedure in different depths of sediments from Sungai Buloh, Selangor, Malaysia. *Journal of Hazardous Materials*, 19: 402-410.

Qiao, Y., Yang, Y., Gu, J., and Zhao, J. 2013. Distribution and geochemical speciation of heavy metals in sediments from coastal area suffered rapid urbanization, a case study of Shantou Bay, China. *Marine Pollution Bulletin*, 68(1-2): 140–146. doi:10.1016/j.marpolbul.2012.12.003.

Sarkar S. K., Paulo J.C.F, Dibyendu R and Satpathy K.K. 2014. Geochemical speciation and risk assessment of heavy metals in soils and sediments, *Environmental Risk Assessment of Soil Contamination, Maria C. Hernandez-Soriano, Intech Open*, DOI: 10.5772/57295.

Siahaya N., Noor A., Sukamto N., and de Voogd N. 2013. A preliminary effort to assign sponge (*Callispogia* sp) as trace metal biomonitor for Pb, Cd, Zn, and Cr, an environmental perspective in Hative gulf waters Ambon. *Advances in Biological Chemistry*, 549-552.

Takarina, N.D., Browne, D.R. and Risk, M.J. 2004. Speciation of heavy metals in coastal sediment of Semarang, Indonesia. *Marine Pollution Bulletin*, 49: 84-874.

Tupan C.I., E.Y. Herawati, D. Arfiati and Aulanni’am. 2013. Heavy metal of Lead (Pb) profile in water, sediment and seagrass (*Thalassia hemprichii*) in Ambon Island. *Proceeding ICGRC Brawijaya University, Feb 7-8th, 2013*.

Turki A J. 2007. Metal Speciation (Cd, Cu, Pb and Zn) in Sediments from Al Shabab Lagoon, Jeddah, Saudi Arabia JKAU: *Marine Science*, 18: 191-210.

Werolangi S., Samawi M.F., Rastina, Tahir A., Faizaland A., and Massinal A. 2016. Bioavailability of Pb and Cu in sediment of Vegetated Seagrass, *Enhalusacoroides*, from Spermonde Island, Makassar, South Sulawesi, Indonesia. *Research Journal of Environmental Toxicology*, 10: 126-134.

Yang J., Cao L., Wang J., Liu C., Huang C., Cai W., Fang H. and Peng X. 2014. Speciation of metals and assessment of contamination in surface sediments from Daya Bay, South China Sea. *Sustainability*, 6: 9096-9113; doi:10.3390/su6129096.