Tetrazine as a general phototrigger to turn on fluorophores†

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Light-activated fluorescence affords a powerful tool for monitoring subcellular structures and dynamics with enhanced temporal and spatial control of the fluorescence signal. Here, we demonstrate a general and straightforward strategy for using a tetrazine phototrigger to design photoactivatable fluorophores that emit across the visible spectrum. Tetrazine is known to efficiently quench the fluorescence of various fluorophores via a mechanism referred to as through-bond energy transfer. Upon light irradiation, restricted tetrazine moieties undergo a photolysis reaction that generates two nitriles and molecular nitrogen, thus restoring the fluorescence of fluorophores. Significantly, we find that this strategy can be successfully translated and generalized to a wide range of fluorophore scaffolds. Based on these results, we have used this mechanism to design photoactivatable fluorophores targeting cellular organelles and proteins. Compared to widely used phototriggers (e.g., o-nitrobenzyl and nitrophenethyl groups), this study affords a new photoactivation mechanism, in which the quencher is photodecomposed to restore the fluorescence upon light irradiation. Because of the exclusive use of tetrazine as a photoquencher in the design of fluorogenic probes, we anticipate that our current study will significantly facilitate the development of novel photoactivatable fluorophores.

Introduction

Photoactivatable fluorophores, also called turn-on fluorophores,1–7 have numerous advanced biological applications, including detection and release of ions8–11 and metabolites,12–15 monitoring of enzyme activity,16–19 and multiple types of specialized microscopy.7,20–22 Most of these photoactivatable fluorophores are masked by a reactive “cage” group intimately attached to the fluorophore to alter its photophysical properties or to serve as an energy-transfer agent for use in a fluorescence resonance energy transfer (FRET) pair.23 Upon light or chemical treatment, these “caged” fluorophores undergo chemical reactions that release the “cage” groups to regenerate the fluorophores in their active forms.27–35 The timing and site of fluorescence restoration are therefore controlled by the application of the “de-caging” treatment. For two reasons, tetrazine has recently emerged as a very effective quenching group in the design of fluorogenic fluorophores. First, tetrazine absorbs visible light at around 500–550 nm, making it an ideal quencher for a series of fluorophores that are useful for FRET studies.36–39 Second, tetrazine derivatives can undergo an Inverse Electron Demand Diels–Alder (IEDDA) cycloaddition reaction with strained hydrophobic alkenes, such as trans-cyclooctene,40 cyclopropenes,41 or norbornene.42 These bio-orthogonal reactions are characterized by fast reaction rates and high selectivity. By taking advantage of this dual functionality, tetrazine has been conjugated via a flexible linker to rhodamine,39 boron dipyrromethene (BODIPY) and Oregon Green47 and then used as a quencher moiety through the FRET mechanism. Upon destruction of the tetrazine core with strained alkenes in the IEDDA reaction, fluorescence enhancement up to 20-fold has been observed for these fluorophores.37

To further improve the quenching efficiency, recent studies have used tetrazine to quench fluorophore fluorescence via through-bond energy transfer (TBET),48–49 a mechanism that does not require spectral overlap between donors and acceptors.50 In this context, tetrazine was conjugated to fluorophore donors using a conjugated linker that allows for faster energy transfer through bonds relative to nonradiative decay processes. The resulting “caged” fluorophores exhibited superb fluorescence turn-on upon reaction with trans-cyclooctene,45 allowing for enhanced imaging of cellular components due to reduced background fluorescence.
Despite the potential for broad applications of tetrazine-based fluorogenic fluorophores, large and strained hydrophobic alkenes have been exclusively used as chemical triggers for restoring fluorescence. Compared to decaging using chemical triggers, photo-decaging offers superior spatial and temporal control and is generally considered to be more biocompatible. Tetrazine can undergo photolysis, generating relatively inert, unobtrusive nitriles and molecular nitrogen. Gaseous tetrazine is known to absorb visible light that permits temporal control and is generally considered to be more fluorogenic than decaging using methyl-phototriggers. methyl-p-phenylphenyl (Tz-BODIPY, 1) as described previously. The fluorescence quantum yield for 1 (φ < 0.01) in acetonitrile is much lower than that for the nitrile counterpart 2 (φ = 0.60) (Table 1), suggesting the possibility that the photolysis of tetrazine could be used to induce fluorogenic activation of Tz-BODIPY. Indeed, following 20 minute irradiation with 254 nm light (1400 μW cm⁻²), the fluorescence intensity of Tz-BODIPY increased dramatically. The observed 178-fold fluorescence enhancement at 509 nm is similar to the 340-fold fluorescence increase reported in the literature after chemical triggering with trans-cyclooctene (Fig. 2).

The photolysis product was further analyzed by ¹H NMR and ESI-MS analysis. Upon light irradiation, the aromatic protons of Tz-BODIPY exhibited an upfield shift. Signals corresponding to phenyl and methyl protons at 8.68, 7.65, 6.12, 3.04, 2.51 and 1.47 ppm gradually disappeared, and new signals appeared at 7.88, 7.56, 6.11, 2.49 and 1.37 ppm (Fig. 2A). These new signals were identical to those observed with synthesized p-cyanophenyl-BODIPY (CN-BODIPY, 2, Fig. 2). ESI-MS analysis of the reaction product yielded an observed mass of 350.16, in agreement with the calculated mass of CN-BODIPY ([M + H⁺] = 350.15, Fig. S1†). These data, therefore, demonstrate that the photolysis product of Tz-BODIPY is its nitrile counterpart, CN-BODIPY.

**Design of photoactivatable fluorophores using the tetrazine phototrigger**

To demonstrate that the tetrazine motif can be used as a general phototrigger, methyl-s-tetrazine-coumarin (Tz-Coumarin, 3) and methyl-s-tetrazine-silicon rhodamine (Tz-Si-Rhodamine, 5) were prepared (Fig. 1). Next, we characterized the spectroscopic and photochemical properties of these fluorophores with the tetrazine phototrigger and their photolysis products using UV-vis and fluorescence spectroscopies (Table 1 and Fig. S2†). As shown in Table 1, the introduction of the tetrazine phototrigger in all the tested fluorophores led to significant reductions in quantum yield. To assess the efficiency of photoactivation for the tetrazine phototrigger, UV-vis and fluorescence spectroscopy were used to measure the quantum yield of the fluorophores before and after light irradiation. The results are summarized in Table 1.

**Table 1 Photophysical data of tetrazine-caged fluorophores and their nitrile derivatives**

| Compound number | Compound       | λ_{abs} (nm) | λ_{em} (nm) | ε × 10⁴ (M⁻¹ cm⁻¹) | Φ_{MOCN}  | Fluorescence increase |
|-----------------|----------------|--------------|-------------|---------------------|-----------|----------------------|
| 1               | Tz-BODIPY      | 490          | 509         | 1.9                 | 0.004a    | 178-Fold             |
| 2               | CN-BODIPY      | 492          | 509         | 3.6                 | 0.60b     | —                    |
| 3               | Tz-Coumarin    | 347          | 416         | 1.8                 | -0.001b   | 86-Fold              |
| 4               | CN-Coumarin    | 351          | 417         | 1.9                 | 0.19b     | —                    |
| 5               | Tz-Si-Rhodamine| 651          | 671         | 2.6                 | 0.011c    | 35-Fold              |
| 6               | CN-Si-Rhodamine| 652          | 671         | 2.4                 | 0.34d     | —                    |
| 7               | Tz-BODIPY-MOR  | 495          | 511         | 2.6                 | 0.03e     | 45-Fold              |
| 8               | Tz-BODIPY-Ts   | 492          | 510         | 1.8                 | 0.02f     | 36-Fold              |
| 9               | Tz-BODIPY-TPP  | 494          | 509         | 3.0                 | 0.007g    | 97-Fold              |
| 10              | Tz-BODIPY-Halo | 491          | 509         | 2.4                 | 0.002h    | 141-Fold             |

a Fluorescein in 0.1 M NaOH. b Quinine sulfate in 0.5 M H₂SO₄ was used as a reference for measuring the quantum yields. c Rhodamine B in water was used as a reference for measuring the quantum yields. d 254 nm light activation (1400 μW cm⁻²). e Measured after irradiation of compound 5 with 254 nm light.

**Results and discussion**

**Photoactivation of BODIPY fluorophores using the tetrazine phototrigger**

To test whether the tetrazine motif can be incorporated into fluorophores to generate photoactivatable fluorophores, we synthesized the methyl-s-tetrazine-BODIPY conjugate (Tz-BODIPY, 1, Fig. 1) as described previously. The fluorescence quantum yield for 1 (φ < 0.01) in acetonitrile is much lower than that for the nitrile counterpart 2 (φ = 0.60) (Table 1), suggesting the possibility that the photolysis of tetrazine could be used to induce fluorogenic activation of Tz-BODIPY. Indeed, following 20 minute irradiation with 254 nm light (1400 μW cm⁻²), the fluorescence intensity of Tz-BODIPY increased dramatically. The observed 178-fold fluorescence enhancement at 509 nm is similar to the 340-fold fluorescence increase reported in the literature after chemical triggering with trans-cyclooctene (Fig. 2).
fluorescence spectra were recorded after different irradiation times (Fig. S3†). To our delight, the fluorescence intensity of the Tz-Coumarin and Tz-Si-Rhodamine molecules increased by 86 and 35-fold, respectively, suggesting that the introduction of a tetrazine phototrigger could be a general strategy for preparing photoactivatable fluorophores with diverse structures.

Fluorescence imaging using Tz-BODIPY derivatives

Having established the excellent photoactivation properties of the tetrazine phototrigger, we sought to explore the utility of the tetrazine phototrigger for biological imaging. Because of its exceptional turn-on properties, we have used the Tz-BODIPY derivatives in the following study. Ultraviolet light has limited applications as a biological initiator, due to its poor tissue penetration and high phototoxicity. To investigate whether light with longer wavelengths can initiate the tetrazine photolysis reaction, we irradiated Tz-BODIPY with light at 254 nm, 365 nm, and 405 nm, respectively. Gratifyingly, significant fluorescence enhancement was observed upon irradiation at 405 nm, a wavelength that is commonly available in commercial microscopes (Fig. S4†). This wavelength was therefore used for activation in the biological imaging study described below. For this study, it was also important to examine the cellular uptake and intracellular photoactivation of Tz-BODIPY derivatives. A431 cells were incubated with Tz-BODIPY, and intracellular photoactivation was monitored by confocal laser scanning microscopy. As shown in Fig. 3A, while negligible fluorescence was observed before photoactivation of Tz-BODIPY, a 14-fold increase in intracellular green fluorescence was observed after irradiation with 405 nm laser light, suggesting not only good photoactivation but also good cell permeability of Tz-BODIPY. However, Tz-Coumarin and Tz-Si-Rhodamine did not exhibit high fluorescence turn-on in A431 cells due to the high autofluorescence and the possible background fluorescence from the open spiroester form before the irradiation process (Fig. S5†).

Synthesis of organelle-targeting Tz-BODIPYs and their applications in fluorescence imaging

The combination of subcellular targeting and photodecaging has provided a platform for precise control of site-specific molecular release. Accordingly, we explored the possibility of combining a tetrazine phototrigger with subcellular targeting moieties. To stably conjugate organelle-targeting moieties to Tz-BODIPY without affecting the its photochemical properties, we developed a robust synthetic approach for functionalizing the BODIPY molecule (Fig. 4). In brief, an aldehyde moiety was first introduced at position 2 of BODIPY using the Vilsmeier–Haack reaction, followed by a Pinnick oxidation to yield carboxylic acid. The resulting carboxyl-functionalized BODIPY was obtained in good yield and coupled to amine-bearing organelle-targeting groups using N-[[(dimethylamino)-1H-1,2,3-triazo[4,5-b]pyridin-1-ylmethylene]-N-methylmethanaminium hexafluorophosphate-N-oxide (HATU)-mediated amide coupling to yield organelle-targeting Tz-BODIPYs. Using this protocol, we have prepared Tz-BODIPY with the following organelle-directing groups: phenyl sulfonamide (Tz-BODIPY-Ts, endoplasmic reticulum, and lysosomes targeting, Fig. 4), morpholine (Tz-BODIPY-MOR, lysosome targeting, Fig. 4), and triphenylphosphonium (Tz-BODIPY-TPP, mitochondria targeting, Fig. 4). To our delight, all three organelle-targeting Tz-BODIPYs were stable in the absence of light (Fig. S6†) and exhibited significant fluorescence enhancement.
after light irradiation in acetonitrile (Table 1, Fig. S7†). In order to evaluate the localization of these organelle-targeting Tz-BODIPY derivatives, confocal laser scanning microscopy was carried out on A431 cells. Commercial MitoView™ 633, ER-Tracker™ Blue-White DPX, and LysoView™ 633 were used as markers for mitochondria, endoplasmic reticulum, and lysosomes, respectively. All three BODIPY compounds had negligible fluorescence prior to irradiation with the 405 nm laser, but exhibited distinct increases in fluorescence signals following tetrazine triggering. Turn-on ratios of 12, 3.5 and 7 were observed for Tz-BODIPY-TPP 9, Tz-BODIPY-MOR 7, and Tz-BODIPY-Ts 8, respectively (Fig. 3B).

Most importantly, good co-localization was observed between the green fluorescence of the BODIPY compounds and the red fluorescence of the corresponding organelle-targeting commercial dyes (Fig. 3B). This is reflected by the high Pearson’s correlation coefficients found for Tz-BODIPY-Ts (γ = 0.85), Tz-BODIPY-MOR (γ = 0.84), and Tz-BODIPY-TPP (γ = 0.75). These data demonstrate that the photoactivity of Tz-BODIPYs can be productively combined with the specificity of subcellular targeting molecules for organelle imaging.

Super-resolution imaging using tetrazine-caged BODIPY fluorophores

Photoactivatable fluorescent probes should find useful applications in super-resolution imaging (e.g., Photoactivated Localization Microscopy, PALM; Stochastic Optical Reconstruction Microscopy, STORM), a technology that allows imaging beyond the diffraction limit. Thus, we explored the possibility of using Tz-BODIPY labeling technology, which is based on covalent bond formation between HaloTag-fused proteins and a synthetic HaloTag ligand. The HaloTag tetrazine-BODIPY conjugate (Fig. 5A) was synthesized following a strategy similar to that used for the organelle-targeting probes. The fluorescence quantum yield of the compound (Φ = 0.002, Table 1) was in the range of that of the other synthesized Tz-BODIPY dyes. After 40 minutes of 254 nm light irradiation using a handheld UV lamp, a 141-fold fluorescence enhancement was observed (Table 1 and Fig. S7†), while no significant fluorescence increase was detected in the absence of light (Fig. S8†). Next, CHO-K1 cells expressing histone 2B (H2B)-HaloTag were incubated with Tz-BODIPY-Halo 10 for 30 min and then incubated in fresh culture medium for another 30 min. Intracellular photoactivation was initially monitored via confocal laser scanning microscopy. As shown in Fig. S9†, an 8-fold increase in intracellular fluorescence was observed after 405 nm laser irradiation. Co-incubation of compound 10 with the commercial nuclear dye, DRAQ5, confirmed a distinct nuclear localization pattern for the histone conjugate. Subsequently, we utilized Tz-BODIPY-Halo 10 for PALM imaging of histone-2B in CHO-K1 cells by activating Tz-BODIPY-Halo with the 405 nm laser contained in the microscope (Fig. S9†). By capturing sets of 20 000–30 000 imaging frames, we reconstructed the super-resolution image of H2B localization (Fig. 5B). Compared to conventional widefield fluorescence imaging, the reconstructed PALM imaging exhibits much better resolution along with high molecular accuracy (ca. 40.8 nm) (Fig. 5C). It is worth mentioning that this PALM imaging can be performed under physiological conditions without the need for cytotoxic reducing buffers to induce the fluorescence on-off transition cycles. Taken together, our data demonstrate that the tetrazine-caged BODIPY fluorophores developed in this study are compatible with existing protein labeling technologies and can be used for super-resolution imaging of living cells.

Conclusions

In summary, we have designed and characterized a unique class of photoactivatable tetrazine-based probes in which the tetrazine moiety serves as both a fluorescence quencher and a phototrigger. The utility of these turn-on probes is further demonstrated by our success with organelle imaging and super-resolution imaging under physiological conditions that do not require the use of a cytotoxic imaging cocktail. We are currently exploring the use of a longer wavelength two-photon laser to dissociate the tetrazine moiety, as this process has been reported for tetrazine by measuring the evolution of HCN after irradiation at 492 and 532 nm. In contrast to the widely used phototriggers (e.g., o-nitrobenzyl and nitrophenethyl groups) altering the electron
structure of fluorescent molecules upon light irradiation,\textsuperscript{78-80} this study affords a new photoactivation mechanism, in which the TBET quencher is photodecomposed to restore the fluorescence. Besides serving as a TBET quencher, the tetrazine moiety has also been exclusively used as a FRET acceptor to design FRET-based fluorogenic probes. Thus, we believe that this work will open up the possibility of using tetrazine as a photo-trigger to activate other FRET-based fluorogenic probes.

Conflicts of interest

There are no conflicts to declare.

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