1,2-Dimethylindole-3-sulfonyl (MIS) as protecting group for the side chain of arginine

Albert Isidro\textsuperscript{a}, Daniel Latassa\textsuperscript{b}, Matthieu Giraud\textsuperscript{b}, Mercedes Álvarez\textsuperscript{accd} and Fernando Albericio\textsuperscript{ace}

\textsuperscript{a}Institute for Research in Biomedicine, Barcelona Science Park, Baldiri Reixac 10, 08028 Barcelona, Spain. E-mail: albericio@irbbarcelona.org, mercedes.alvarez@irbbarcelona.org; Fax: +3493 4037126; Tel: +3493 4037088
\textsuperscript{b}Lonza AG., TIDES, CH-3930, Visp, Switzerland
\textsuperscript{c}CIBER-BBN, Networking Centre on Bioengineering, Biomaterials and Nanomedicine, Barcelona Science Park, Baldiri Reixac 10, , 08028 Barcelona, Spain
\textsuperscript{d}Laboratory of Organic Chemistry, Faculty of Pharmacy, University of Barcelona, 08028 Barcelona, Spain
\textsuperscript{e}Department of Organic Chemistry, University of Barcelona, Martí i Franqués 1, 08028 Barcelona, Spain

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The protection of arginine (Arg) side chains is a crucial issue in peptide chemistry because of the propensity of the basic guanidinium group to produce side reactions. Currently, sulfonyl-type protecting groups, such as 2,2,5,7,8-pentamethylchroman (Pmc) and 2,2,4,6,7-pentamethyldihydrobenzofuran (Pbf), are the most widely used for this purpose. Nevertheless, Arg side chain protection remains problematic as a result of the acid stability of these two compounds. This issue is even more relevant in Arg-rich sequences, acid-sensitive peptides and large-scale syntheses. The 1,2-dimethylindole-3-sulfonyl (MIS) group is more acid-labile than Pmc and Pbf and can therefore be a better option for Arg side chain protection. In addition, MIS is compatible with tryptophan-containing peptides.

Introduction

Most peptides synthesized on a solid-phase are prepared using the Fmoc/tert-butyl strategy.\textsuperscript{1,2} Thus, \(\alpha\)-amino temporary protection is achieved with the base labile 9-fluorenylmethoxycarbonyl (Fmoc) group; amino acid side chains are protected by trifluoroacetic acid (TFA)-labile protecting groups, usually \(\text{tBu}\) derivatives; and the C-terminal amino acid is anchored to the solid support through a TFA-labile linker/handle. Nevertheless, tert-butyl-type protection of a number of amino acids is not the best option because of factors such as inefficiency in preventing side reactions or inadequate TFA lability. Among these amino acids, protection of the basic guanidinium group of Arginine (Arg) is possibly the most critical case.\textsuperscript{3}

Currently, the most frequently used TFA-labile Arg-protecting groups are based on electron-
rich benzene sulfonyl moieties. These groups are, in increasing order of acid lability: 4-methoxy-2,6-dimethylbenzenesulfonyl (Mds), 4-methoxy-2,3,6-trimethylsulfonyl (Mtr), 2,2,5,7,8-pentamethylchroman-6-sulfonyl (Pmc), and 2,2,4,6,7-pentamethyldihydrobenzofurane-5-sulfonyl (Pbf) (Fig. 1).

All of these mask the reactivity of the \( N^\text{O} \), are commercially available and have been extensively used in the Fmoc/Bu solid-phase strategy. Nevertheless, the problem of the side chain protection of Arg remains unsolved because even the Pbf group is too stable to TFA and its removal requires high TFA concentrations and long treatment times, which may not be appropriate for acid-sensitive peptides. The conditions become increasingly more demanding when preparing multiple Arg-containing peptides, which show biological properties of great interest.

The design of a new sulfonyl-based Arg-protecting group is not a straightforward process in the sense of simply adding electron-donating groups to an aromatic ring, because the planarity of the system, which is essential for TFA lability, is not easily conserved because of the presence of the sulfonyl group. Thus, trimethoxybenzenesulfonyl (Mtb), which contains more electron-rich substituents (3 MeO) is less acid-labile than Mds (1 MeO, 2 Me) and Mtr (1 MeO, 3 Me). This characteristic is attributed to the loss of planarity caused by the presence of the two methoxy groups near the sulfonyl group. Furthermore, the sulfonyl derivative of 3,4-ethylenedioxythiophene (EDOT), whose derived compounds are highly labile to TFA as carboxylic acid protectors, is not labile as an Arg side-chain protector, possibly because of the same loss of planarity. Common side-reactions associated with the use of these benzenesulfonyl-based protecting groups are arylation of sensitive residues, such as Trp, or sulfonation of Trp and/or Arg residues themselves. These side reactions are favored by the decomposition of the sulfonyl-protecting group in two moieties, the arylcarbocation, which is an alkylating agent, and the sulfonyl part, which can cause addition of sulfur trioxide on the peptidic chain.

In an attempt to overcome the above mentioned drawbacks, we describe herein a new more acid-labile Arg side chain-protecting group based on the indole system.

Results and discussion

General

A TFA-labile protecting group should be based on an electron-rich system. In this regard, \( N \)-alkylindole derivatives have been used as acid-labile amide linkers and amide backbone
protectors. Taking this into account, we chose 1,2-dimethylindole-3-sulfonyl (MIS) as a guanidinium-protecting group (Fig. 1). The extra methyl at position 2 should increase the acid lability of the protecting group and prevent electrophilic aromatic substitution. Furthermore, the 1,2-dimethylindole is commercially available.

As the 1,2-dimethylindole is prone to polymerization in strong acidic conditions, sulfonation of the indole ring must be carried out in neutral or basic media. Thus, chlorosulfonic acid, which is the reagent of choice for Pmc and Pbf sulfonylation, cannot be used in the case of 1,2-dimethylindole. Nevertheless, the use of sulfur trioxide pyridine complex yielded the corresponding pyridinium sulfonate in good yield. Chlorination under mild conditions by treatment with oxalyl chloride yielded 1,2-dimethylindole-3-sulfonyl chloride (MIS-Cl). These conditions gave much better overall yield (80%) to those attained with Pbf and Pmc (51% and 53% respectively), with the advantage that 1,2-dimethylindole is commercially available (Scheme 1).

**Synthesis of multiple arginine-containing peptides using MIS and Pbf protection**

We prepared Fmoc-Arg(MIS)-OH in a similar way to Pmc/Pbf derivatives, using Z-Arg-OH as starting material. Z-Arg-OH was sulfonlated at the N\(^{\text{N}}\) position with MIS-Cl and the Z group was removed via catalytic hydrogenolysis. Final Fmoc protection was achieved by using Fmoc-2-mercaptobenzothiazole (Fmoc-2-MBT) because the use of other more active Fmoc derivatives leads to the formation of dipeptides or other side reactions.

As Pbf removal is more complicated in multiple Arg-containing peptides, Ac-Phe-Arg-Arg-Arg-Arg-Val-NH\(_2\) was chosen as a model peptide to compare the acid lability of MIS and Pbf. The corresponding Pbf- and MIS-protected peptides were prepared using standard solid-phase peptide synthesis protocols on Sieber amide resin, which allows cleavage from the resin with small amounts of TFA (2%), thereby yielding the MIS- and Pbf-protected peptides respectively with excellent purity.
Removal assays

To compare the acid lability of the Pbf group, which is more acid-labile than Pmc, with the MIS group, protected peptide-bonded resins were treated with TFA–CH₂Cl₂–H₂O–TIS (50 : 45 : 2.5 : 2.5) for 30 (MIS, 100%; Pbf, 4%) and 60 min (MIS, 100%; Pbf, 38%). These assays revealed that the MIS group is considerably more acid-labile than the Pbf one.

Also, the MIS derivative generated in the removal process differs from the case of Pbf. For Pbf and Pmc, 2,2,5,7,8-pentamethylchroman and 2,2,4,6,7-pentamethyldihydrobenzofurane, respectively, are formed via a desulfonylation mechanism, while for MIS, the sulfonic acid (MIS-OH) was stable and was not desulfonated.

Optimization of the scavengers used in the removal

As MIS-OH is a polar compound, it precipitates during the ether treatment after the cleavage step. Alternative scavengers to H₂O were tested to reduce the amounts of the strongly UV absorbant MIS-OH in order to facilitate purification. Among the scavengers tested, the optimum were 10% 3,4-dimethoxyphenol, 1,3,5-trimethoxybenzene (Tmb) or 3,5-dimethoxyphenol. The use of these scavengers reduced the amounts of MIS-OH more than 10 fold (40 times in the case of Tmb), thereby simplifying HPLC purification to yield the final product.

Synthesis of Trp-containing peptides

Scheme 1 Synthetic pathway for the preparation of Fmoc-Arg(MIS)-OH.
To check the compatibility of the MIS group with Trp, we first synthesized the model peptides Z-Arg(MIS)-Trp(Boc)-Ala-Gly-NH₂ and Z-Arg(Pbf)-Trp(Boc)-Ala-Gly-NH₂ on a Sieber amide resin, which were obtained with an excellent HPLC purity. Afterwards, both resins were treated with TFA-CH₂Cl₂-trimethoxybenzene (50:40:10) to compare the purities of Trp-containing peptides after MIS and Pbf removal. Trp alkylation or sulfonation was not detected in either case. The purity of the crude product was greater in the case of MIS and neither the MIS-protected peptides nor MIS-OH were detected by LC-MS. Nevertheless, in the case of the Pbf experiment, considerable amounts of the Pbf-protected peptide were detected (34% compared to unprotected peptide, HPLC, λ = 220 nm).

In summary, MIS is the most acid-labile sulfonyl-type protecting group for Arg described to date. This feature makes it highly convenient for the synthesis of multiple Arg-containing peptides or peptides that contain acid-sensitive moieties. Furthermore, MIS is compatible with Trp-containing peptides.

Experimental section

Synthesis of the protecting group and Arginine protection

Pyridinium 1,2-dimethylindole-3-sulfonate (1). 1,2-Dimethylindole (14.5 g, 99.8 mmol) and sulfur trioxide pyridine complex (19.1 g, 119.8 mmol) were dissolved in pyridine (70 mL) under an Ar atmosphere. The reaction mixture was refluxed for 1 h and HPLC indicated the reaction was complete (99.2% HPLC conversion, 254 nm). The reaction mixture was cooled to 60 °C and concentrated under vacuum to give a solid. The crude product was used directly for the next step.¹²¹H NMR (400 MHz, D₂O): δ = 8.44 (d, 2H, 2CH, J = 5.8 Hz), 8.31 (m, 1H, CH), 7.75 (m, 2H, 2CH), 7.67 (d, 1H, CH, J = 7.7 Hz), 7.14 (d, 1H, CH, J = 7.4 Hz), 7.05 (m, 2H, 2CH), 3.38 (s, 3H, CH₃), 2.41 (s, 3H, CH₃).¹³C NMR (100 MHz, D₂O): δ = 147.0 (CH), 140.9 (CH), 139.2 (C), 135.6 (C), 135.9 (C), 127.3 (CH), 124.1 (C), 122.0 (CH), 121.0 (CH), 119.2 (CH), 112.8 (C), 109.9 (CH), 29.2 (CH₃), 10.4 (CH₃). HRMS (CI): m/z calcd. for C₁₀H₁₀NO₃S [M–H]+ 224.0386, found 224.0388.

1,2-Dimethylindole-3-sulfonyl chloride (MIS-Cl) (2). All the crude product obtained in the previous step was suspended in dry CH₂Cl₂ (200 mL) under Ar atmosphere. The suspension was cooled in an ice bath and oxalyl chloride (20.0 g, 158 mmol) was slowly added. DMF (0.5 mL) was then slowly and carefully added and vigorous effervescence was observed. The reaction mixture was stirred in an ice bath for a further 30 min until the effervescence ceased and was then stirred at room temperature for 2 h. An aliquot (6 L) was then treated with MeOH for 20 min and injected into the HPLC apparatus, which showed the presence of methyl 1,2-dimethylindole-3-sulfonate and an absence of starting material. CH₂Cl₂ (200 mL) was added to the reaction mixture and it was cooled to below 5 °C. H₂O (2–8 °C, 150 mL) was added and the mixture was stirred for 10 min. The organic phase was separated and washed with 2–8 °C H₂O (2 × 150 mL), and dried with anhydrous MgSO₄ (15 g). The solution was then concentrated to ca. 40 mL. The solid was filtered off and washed with CH₂Cl₂–n-hexane (1:1, 60 mL). The solid was dried under vacuum to give light pink solid (19.6 g, 80.4% yield, 98% HPLC purity, base on 1,2-dimethylindole). Mp = 67.7–73.5 °C.¹¹H NMR (400 MHz, DMSO-d₆): δ = 7.82 (d, 1H, CH, J = 7.8 Hz), 7.36 (d, 1H, NH, J = 8.0 Hz), 7.08 (m, 2H, 2CH), 7.00 (m, 2H, 2CH), 3.63 (s, 3H, CH₃), 2.56 (s, 3H, CH₃).¹³C NMR (100 MHz, DMSO-d₆): δ = 137.2 (C), 135.9 (C), 125.5 (C), 121.4 (CH), 120.8 (CH), 120.1 (CH), 109.7 (CH), 30.0 (CH₃), 11.3 (CH₃). HRMS (CI): m/z calcd. for C₁₀H₁₀NO₃S [M–Cl]− 208.0426, found 208.0427.

Z-L-Arg(MIS)-OH (3). Z-L-Arg-OH (2.05 g, 6.7 mmol) was suspended in 3 N aqueous NaOH (6.7 mL, 20 mmol) and acetone (13.3 mL) was added to dissolve the product. The reaction was cooled in an ice bath and 3 N aqueous NaOH (6.7 mL) and a solution of

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compound 2 (3.69 g, 14.7 mmol) in acetonitrile (13.3 mL) were simultaneously added over 10 min. The reaction mixture was stirred at 0 °C for 2 h and at room temperature for a further 2 h. After that time, starting material 2 was no longer detected by TLC (hexane–EtOAc, 1 : 1). H₂O (100 mL) was added and the suspension was washed with diethyl ether (3 × 80 mL). The aqueous phase was acidified to pH 2–3 by addition of 1 N HCl, the precipitate obtained was filtered, washed with acidic water (pH 2–3) and dried in vacuo. The crude product obtained was purified by column chromatography (CH₂Cl₂, MeOH, 1% HOAc). The solvent of the pure fractions was removed in vacuo to yield an oil. This process was repeated. Hexane and CH₂Cl₂ were then sequentially added and a precipitate appeared on scratching. The solvent was decanted and the solid was washed 4 times with CH₂Cl₂–hexane (enough hexane to precipitate all the product) to remove HOAc and give 3 (0.70 g, 20.4% yield).²⁴ Mp = 155.5–159.1 °C.¹H NMR (400 MHz, DMSO-d-6): δ = 7.85 (d, 1H, CH, J = 7.6 Hz), 7.52 (d, 1H, NH, J = 8.0 Hz), 7.43 (d, 1H, CH, J = 8.0 Hz), 7.30 (m, 5H, 5CH Z), 7.10 (m, 2H, 2CH), 5.01 (s, 2H, CH₂), 3.87 (m, 1H, CH), 3.66 (s, 3H, CH₃), 3.0 (m, 2H, CH₂), 2.60 (s, 3H, CH₃), 1.64 (m, 1H, CH₂), 1.49 (m, 1H, CH₂), 1.41 (m, 2H, CH₂).¹³C NMR (100 MHz, DMSO-d-6): δ = 174.4 (C), 157.0 (C), 156.8 (C), 139.4 (C), 137.7 (C), 135.9 (C), 129.0 (CH), 128.5 (CH), 128.4 (CH), 125.2 (C), 121.1 (CH), 120.1 (CH), 110.4 (CH), 66.1 (CH₂), 54.3 (CH), 40.0 (CH₂), 30.2 (CH₃), 28.9 (CH₂), 26.4 (CH₂), 11.4 (CH₃). HRMS (CI): m/z calcd. for C₁₄H₂₄N₅O₆S [M + H]+ 516.1911, found 516.1911.

H-L-Arg(MIS)-OH (4). A mixture of 3 (486 mg, 0.94 mmol) and Pd/C (10%) (110 mg) in MeOH (60 mL) was hydrogenated overnight at atmospheric pressure. After this time, TLC (CH₂Cl₂–MeOH–HOAc, 90:9:1) still showed some starting material. More 10% Pd/C (100 mg) was added and the mixture was hydrogenated for a further 24 h, after which TLC showed the absence of starting material. The reaction mixture was filtered over celite and evaporated to dryness to yield 4 (352 mg, 98% yield). Mp = 153.2–155.0 °C.¹H NMR (400 MHz, DMSO-d-6): δ = 7.83 (d, 1H, CH, J = 7.6 Hz), 7.47 (d, 1H, NH, J = 8.1 Hz), 7.42 (d, 1H, CH, J = 8.1 Hz), 7.11 (m, 2H, 2CH), 3.65 (s, 3H, CH₃), 3.17 (m, 1H, CH), 3.00 (m, 2H, CH₂), 2.60 (s, 3H, CH₃), 1.65 (m, 1H, CH₂), 1.54 (m, 1H, CH₂), 1.42 (m, 2H, CH₂). HRMS (CI): m/z calcd. for C₁₆H₂₄N₅O₆S [M + H]+ 382.1544, found 382.1542.

Fmoc-Arg(MIS)-OH (5). H-Arg(MIS)-OH (250 mg, 0.658 mmol) was suspended in 1% aqueous Na₂CO₃ (2 mL). 1,4-dioxane (2 mL) was added and the product was dissolved. The pH was basified to 9–10 with saturated aqueous Na₂CO₃ (300 μL in our case). Fmoc-2-mercaptobenzotiazole (Fmoc-2-MBT) (256 mg, 0.658 mmol) in 1,4-dioxane (700 L) was added. The pH was adjusted to 9–10 with saturated aqueous Na₂CO₃ and the resulting suspension was stirred overnight. After 14 h of reaction, H₂O (9 mL) was added, the pH was neutralized with 1 N HCl and the solution was washed with tert-butylmethyl ether (3 × 5 mL). The aqueous phase was acidified to pH 2–3 with 1 N HCl and extracted with EtOAc (3 × 7 mL). Note that to dissolve the precipitated product vigorous stirring is required. The organic phases were pooled, dried over dry MgSO₄, filtered and evaporated to dryness, thereby yielding a solid. Various co-evaporations with CH₂Cl₂ were performed to yield the desired product as a solid (207 mg, 52% yield). Mp = 137.2–146.4 °C.¹H NMR (400 MHz, DMSO-d-6): δ = 7.86 (m, 3H, 3CH), 7.70 (d, 2H, 2CH, J = 7.4 Hz), 7.59 (d, 1H, NH, J = 7.9 Hz), 7.42 (d, 1H, CH, J = 8.1 Hz), 7.39 (m, 2H, 2CH), 7.30 (m, 2H, 2CH), 7.10 (m, 2H, 2CH), 4.27 (m, 2H, CH₂), 4.20 (m, 1H, CH), 3.86 (m, 1H, CH), 3.66 (s, 3H, CH₃), 3.01 (m, 2H, CH₂), 2.61 (s, 3H, CH₃), 1.65 (m, 1H, CH₂), 1.52 (m, 1H, CH₂), 1.38 (m, 2H, CH₂).¹³C NMR (100 MHz, DMSO-d-6): δ = 174.4 (C), 157.0 (C), 156.8 (C), 144.5 (C), 141.4 (C), 139.4 (C), 128.3 (CH), 127.8 (CH), 126.0 (CH), 125.2 (C), 122.1 (CH), 121.1 (CH), 120.8 (CH), 120.1 (CH), 110.4 (CH), 66.3 (CH₂), 55.6 (CH), 47.3 (CH), 40.0 (CH₂), 30.2 (CH₃), 28.8 (CH₂), 26.5 (CH₂), 11.4 (CH₃).
Synthesis of arginine-containing model peptides using MIS and Pbf protection

Ac-Phe-Arg-Arg-Arg-Arg-Val-NH$_2$ (model peptide 1). Ac-Phe-Arg(MIS)-Arg(MIS)-Arg(MIS)-Arg(MIS)-Arg(MIS)-Val-NH$_2$ Sieber amide resin (25 mg, 0.42 mmol/g) was placed in a 2 mL polypropylene syringe fitted with a polyethylene filter disc. The resin was swollen with CH$_2$Cl$_2$, washings with CH$_2$Cl$_2$ and DMF were carried out and the Fmoc group was removed by treatment with piperidine–DMF (2 : 8) (1 × 1 min, 2 × 10 min). Fmoc-L-Val-OH (14.3 mg, 42.1 μmol) was coupled using HOBt (5.7 mg, 42.1 μmol) and DIC (6.7 μL, 42.1 μmol) in DMF, t = 90 min. The Fmoc group was removed in the usual way and Fmoc-L-Arg(MIS)-OH (15.8 mg, 26.3 μmol) was coupled using PyBOP (13.7 mg, 26.3 μmol) HOAt (3.6 mg, 26.3 μmol) and DIPEA (13.4 μL, 78.9 μmol) in DMF for 90 min. The resin was acetylated by treatment with Ac$_2$O (50 eq.) and DIPEA (50 eq.) in DMF for 25 min. The Fmoc group was removed and the same procedure was repeated three more times, acetyling the resin before each Fmoc removal. After the last Fmoc removal, Fmoc-L-Phe (13.6 mg, 35 μmol) was coupled using PyBOP (18.3 mg, 35 μmol) HOAt (4.8 mg, 35 μmol) and DIPEA (17.9 μL, 105.2 μmol) in DMF for 90 min. After the cleavage time, the solution was poured into H$_2$O (4 mL), and TFA and CH$_2$Cl$_2$ were evaporated. The resulting aqueous solution was washed with CH$_2$Cl$_2$ (6 × 1 mL), frozen, lyophilized and analyzed by HPLC (λ = 220 nm) and ESMS or MALDI-TOF.

HRMS (CI): m/z calcd. for C$_{31}$H$_{34}$N$_5$O$_6$S [M + H]$^+$ 604.2224, found 604.2222.

Ac-Phe-Arg(Pbf)-Arg(Pbf)-Arg(Pbf)-Arg(Pbf)-Val-NH$_2$. The same procedure as for the synthesis of peptide 1 was used but replacing Fmoc-L-Arg(MIS)-OH by Fmoc-L-Arg(Pbf)-OH (17.1 mg, 26.3 μmol). The product obtained was characterised by LC-MS and HRMS (CI): m/z calcd. for C$_{122}$H$_{136}$N$_{19}$O$_{19}$S$_4$ [M + Na]$^+$ 1938.9137, found 1938.9202.

Removal assays. General procedure: the resin (3 mg) was treated with cleavage solution (50 μL). After the cleavage time, the solution was poured into H$_2$O (4 mL), and TFA and CH$_2$Cl$_2$ were evaporated. The resulting aqueous solution was washed with CH$_2$Cl$_2$ (6 × 1 mL), frozen, lyophilized and analyzed by HPLC (λ = 220 nm) and ESMS or MALDI-TOF.

Optimization of the scavengers. The same procedure as for the removal assays was followed. In all the experiments the resin was treated with TFA-CH$_2$Cl$_2$-scavenger (50 : 40 : 10) (50 μL) for 1 h. The scavengers tested were 3,4-dimethoxyphenol, 1,3,5-trimethoxybenzene (Tmb) or 3,5-dimethoxyphenol.

Z-Arg-Trp-Ala-Gly-NH$_2$ (model peptide 2). Z-Arg(MIS)-Trp(Boc)-Ala-Gly-NH$_2$ and Z-Arg(Pbf)-Trp(Boc)-Ala-Gly-NH$_2$. Sieber amide resin (70 mg, 0.40 mmol/g) was placed in a 2 mL polypropylene syringe fitted with a polyethylene filter disc. The resin was swollen with CH$_2$Cl$_2$, washings with CH$_2$Cl$_2$ and DMF were carried out and the Fmoc group was removed. Fmoc-L-Gly-OH (33.3 mg, 112 μmol), Fmoc-L-Ala-OH (34.9 mg, 112 μmol) and Fmoc-L-Trp (Boc)-OH (59.0 mg, 112 μmol) were sequentially coupled using PyBOP (58.3 mg, 112 μmol) HOAt (15.2 mg, 112 μmol) and DIPEA (57.4 μL, 336 μmol) in DMF, t = 1.5 h. The resin was divided into two equal parts.
Part 1 [Z-Arg(MIS)-Trp(Boc)-Ala-Gly-NH₂]. Z-Arg(MIS)-OH (28.9 mg, 56 μmol) was coupled using PyBOP (29.2 mg, 56 μmol) HOAt (7.6 mg, 56 μmol) and DIPEA (28.7 μL, 168 μmol) in DMF, t = 1.5 h. The resin was washed with DMF, CH₂Cl₂ and diethyl ether, dried in vacuo and divided into 4 mg aliquots. One of them was swollen with CH₂Cl₂ and treated with 1.5 mL of TFA–CH₂Cl₂–TIS–H₂O (2 : 93 : 2.5 : 2.5) for 20 min in order to cleave the protected peptide from the resin. The resin was filtered and the collected solution was diluted with CH₂Cl₂ and neutralised by adding DIPEA (80 μL, 1.2 eq. per eq. of TFA). The solvent was then removed in vacuo, and H₂O and AcCN were added and the solution was frozen and lyophilized. The product obtained was characterised by LC-MS (95% purity). HRMS (Cl): m/z calcd. for C₄₅H₅₇N₁₀O₁₀S [M + H]+ 929.3974, found 929.3969.

Part 2 [Z-Arg(Pbf)-Trp(Boc)-Ala-Gly-NH₂]. Fmoc-Arg(Pbf)-OH (36.3 mg, 56 μmol) was coupled using PyBOP (29.2 mg, 56 μmol), HOAt (7.6 mg, 56 μmol) and DIPEA (28.7 μL, 168 μmol) in DMF, t = 1.5 h. The Fmoc group was removed and the free amine was protected with the Z group by treatment with Z-OSu (14.0 mg, 56 μmol) and DIPEA (35.9 μL, 210 μmol). The resin was then washed with DMF, CH₂Cl₂ and diethyl ether, dried in vacuo, and divided into 4 mg aliquots, one of which was cleaved in the same way as for Part 1. The product obtained was characterised by LC-MS (96% purity).

Z-Arg-Trp-Ala-Gly-NH₂ from Z-Arg(MIS)-Trp(Boc)-Ala-Gly-NH₂. Two aliquots from Part 1 were treated with TFA–CH₂Cl₂–1,3,5-trimehtoxybenzene (50 : 40 : 10) and TFA–CH₂Cl₂–H₂O (50 : 45 : 5) respectively for 1 h following the General procedure for the removal assays described above. In the latter case, no CH₂Cl₂ washings were performed. The two crude products resulting from these treatments were analyzed by LC-MS. No Trp alkylation or sulfonation nor MIS-protected peptide were observed.

Z-Arg-Trp-Ala-Gly-NH₂ from Z-Arg(Pbf)-Trp(Boc)-Ala-Gly-NH₂. An aliquot from Part 2 was treated with TFA-CH₂Cl₂-trimehtoxybenzene (50 : 40 : 10) for 1 h following the General procedure for the removal assays described above. The target peptide was analyzed by LC-MS (60% purity). 17% of Pbf- protected peptide was detected and no Trp alkylation or sulfonation was observed.

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4 ω,ω'-bis-Boc Arginine derivatives have a similar acid lability to TFA as Pbf but are not so widely used. The coupling rates of bis-Boc protected Arg are low. B. F. Lundt, N. L.
The use of the -NO$_2$ protected arginine derivative (M. Bergmann, L. Zervas and H. Rinke, H-S Z. Physiol. Chem, 1934, 224, 40) which is also commercially available, has the main drawback of long reaction times, which is an inconvenient in the case of sensitive peptides. For instance in the case of hydrogenolysis partial hydrogenation of Trp or even Phe can occur G. T. Young, D. J. Schafer, D. F. Elliott and R. Wade, J. Chem. Soc. C., 1971, 46 as well as the 10,11-dihydro-5H-dibenzo[a,d]cyclohepten-5-yl [5-dibenzosuberyl] (Sub) (M. Noda and M. Kiffe, J. Pept. Res., 1997, 50, 329), have been also proposed but not broadly used.

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In other experiments the final product was isolated using the following work up: The reaction mixture was cooled to room temperature and H$_2$O was added (400 mL). The resulting solution was washed with diethyl ether (4 × 250 mL). The aqueous phase was evaporated to dryness and dried in the vacuum dessicator to render a red oily solid (96% yield), which was characterized by NMR and HRMS.

A similar procedure which avoids the column chromatography purification and may be more convenient for industrial scale has also been developed: Z-L-Arg-OH (5.0 g, 16.2 mmol) was dissolved in acetone (160 mL) and 3 N aqueous NaOH (45 mL, 135 mmol). The reaction was cooled in an ice bath and compound 2 (3.85 g, 15.8 mmol) dissolved in acetone (100 mL) was added over 10 min. The reaction mixture was stirred for 1 h at 0 °C. Additional, 2 (1.9 g, 7.8 mmol) in acetone (50 mL) was then added followed by 90 min of stirring at 0 °C. Finally, a last portion of compound 2 (1.9 g, 7.8 mmol) in acetone (50 mL)
was added and the reaction mixture was stirred at 0 °C for additional 30 min and at room
temperature for 3 h. At this point, species 2 was no longer observed by HPLC. Acetone,
was distilled out under vacuum and H₂O (100 mL) was added (pH = 14). Then EtOAc (150
mL) was added and the pH adjusted to 3.0 with 50% aqueous citric acid. The solution
became 3 phases. The above organic phase as well as the underneath oil phase were
separated. H₂O (30 mL) was added to the oil phase, and the pH adjusted to 10. To make the
solution more homogeneous, EtOAc (50 mL) was added and the pH adjusted to 3.0. The
EtOAc phase was separated and combined with the EtOAc phase from the first extraction.
The combining solution was concentrated to 100 mL. H₂O (100 mL) was added, the pH
was adjusted firstly to 10 and finally to 3.0. At this point, the aqueous phase was
discharged. This process was repeated and the EtOAc solution was concentrated to give a
light yellow solid (2.9 g, yield: 24.5%, HPLC purity: 70.6%).

Footnote
† Electronic supplementary information (ESI) available: Detailed experimental procedures for intermediates,
final Fmoc-Arg(MIS)-OH, and peptides; 'H- and 13C-NMR spectra and HPLC chromatograms. See DOI:
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