Original article

Subacute toxic effects of silver nanoparticles oral administration and withdrawal on the structure and function of adult Albino Rats’ hepatic tissue

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A B S T R A C T

Products containing Silver nanoparticles (Ag NPs) are becoming vastly used in our daily life. The widespread increased introduction of Ag NPs in many aspects of life has raised researchers’ concerns regarding their safety and toxicity for biological and environmental life in the past few years. The current study aimed to explore the subsequent effects of Ag NPs withdrawal, following short-term oral administration. Eighteen rats were assigned randomly into three groups (control group “1” and Ag NPs treated groups “2” and “3”; 6 animals each). The control group received normal food and tap water while groups 2 & 3 received 0.5 ml of a solution containing 25 ppm Ag NPs for 14 days. Group 2 rats were sacrificed on day 14 whereas group 3 was left for another 14 days of particle cessation followed by euthanasia on day 28. Functional assessment was done by liver enzyme assays, hydrogen peroxide activity, hepatic Bdnf expression, and P53 immunoreactivity. Hepatic tissue structural assessment was done via hematoxylin and eosin, periodic acid–Schiff as well as Masson’s trichrome stains. The results revealed a significant elevation of Hydrogen peroxide in group 2 only compared to the control group. Hepatic Bdnf and liver enzymes were both insignificantly affected. Structural abnormalities and enhanced apoptosis in hepatic tissue were found 14 days after ceasing the nanoparticles. In conclusion: Structural and functional insults following Ag NPs oral administration continues after particle withdrawal, and interestingly they do not necessitate apparent reflection on liver enzyme assays.

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1. Introduction

Nanotechnology breakthroughs and the invention of nanoparticles (NPs), notably silver nanoparticles (Ag NPs), have made significant contributions to health, drug research, food processing, and agriculture (Yildirim et al., 2011). Nanoparticles are microscopic particles with a diameter of <100 nm. Nanoparticles are more dangerous than particles of a different size. Ag NPs of various sizes can be consumed directly through food, water, cosmetics, and pharmaceuticals (Sardari et al., 2012). In the field of medicine, Ag NPs are used due to their antibacterial, antiviral, and antifungal properties. It is also used as a disinfectant in swimming pools and drinking water (El Mahdy et al., 2014). Furthermore, it was observed in the literature that oral administration of Ag NPs prepared solution as a treatment raised the risk of argyria (Griffith et al., 2015). Argyria is a benign health condition related to the ingestion of Ag which leads to blue-grayish skin discoloration through the
preferential deposition of Ag in the “basal lamina” of different internal soft tissues as the spleen, blood vessels, liver, gastrointestinal tract, and kidney (Griffith et al., 2015; Flores-López et al., 2019).

The hazardous effect of Ag NPs on humans has recently garnered a lot of attention, notably in airborne occupational exposures during nanomaterial manufacturing. Ag NPs’ tiny size enables their transfer from natural barriers such as the lungs, gastrointestinal system, or skin, resulting in acute and chronic harmful consequences (Bakand and Hayes, 2016). Therefore, the widespread introduction of Ag NPs in many aspects of life has raised researchers’ concerns regarding their safety and toxicity for biological and environmental life in the past few years. Consequently, this necessitates a thorough investigation of their dramatically variable interactions at the molecular levels (Akter et al., 2018; Gupta and Xie, 2018). Ag NPs toxicity was proven by the existing data on possible adverse health consequences. AgNPs release silver ions in the presence of water, which cause considerable change in oxidative and immunity responses, as well as a multitude of harmful effects on cells and genes (Akter et al., 2018; Chernousova and Epple, 2013; Cho et al., 2013; Deckers et al., 2008). The form, size, concentration, dose, route of entry into the body, and duration of exposure of Ag NPs, as well as the presence of the aggregated state, are all aspects that can influence Ag NPs toxicity (Ajday et al., 2018; Mao et al., 2018). Therefore, characterization is essential for determining the properties of nanoparticles under investigation (Zhang et al., 2016).

The consumption of various-sized particles through the gastrointestinal tract may have various toxicological implications. However, research on the toxicological effects of nanoparticles on the gastrointestinal tract is lacking. The spleen and liver have previously been identified as key target organs after intravenous injection of several nanomaterials. This is because these organs are part of the reticuloendothelial system (RES), which is responsible for removing foreign substances from the circulatory system (De Jong et al., 2013). Therefore, a more thorough examination of Ag NP’s ability to infiltrate the liver is required. However, a rat liver cell line research proved that 25 ppm of Ag NPs was the most toxic concentration (Akter et al., 2018).

Previous researches suggest that the cytotoxicity effect of Ag NPs is due to the initiation of oxidative stress (Flores-López et al., 2019; Kim et al., 2009; Manke et al., 2013; Rahman et al., 2009). Reactive oxygen species (ROS) are normally produced in cellular reactions. However, when their production rate exceeds the natural cellular antioxidant defense mechanisms, they can lead to an increase in the process of lipid peroxidation, mitochondrial damage, and apoptosis (Flores-López et al., 2019). The enhanced ROS activity in response to Ag NPs leads to production of hydrogen peroxide (H₂O₂) (Patlolla et al., 2015). Therefore, this can provoke the intrinsic apoptotic pathway in the mitochondria via the production of H₂O₂ (Lorenzo et al., 2009; Manke et al., 2013). Brain-derived neurotrophic factor (BDNF) is a neurotrophic factor to help neurons transfer from natural barriers such as the lungs, gastrointestinal system, or skin, resulting in acute and chronic harmful consequences (Bakand and Hayes, 2016). Therefore, in the current study, we are focusing on the sub-acute effects of Ag NPs subacute oral administration and the persistence of harmful effects following the stoppage of Ag NPs oral administration. The oral method was chosen to mimic Ag NP consumption in everyday life such as in the water and food industry and due to the sacrity of researches that used the oral route. The administration of a small dose was conducted to examine the effect of continued daily exposure and how much harm it would cause when it was withdrawn.

2. Methods

2.1. Animals

Healthy adult male Albino Wistar rats (n = 18; 3 groups, 6 rats each; age: 8 ± 1 weeks; 220–225 g) were purchased (Source: The Ophthalmic Research Institute in Giza, Egypt). Female rats were not included in this investigation. This is attributed to the erratic nature of female data resulting from hormonal changes during the female reproductive cycle. Rats were housed in plastic cages in a temperature and humidity-controlled setting (25 2 °C, 55 per cent humidity, and 12 h light/dark cycles). The animals were fed a conventional chow diet (Al-Gomhoureya Co.) that had a carbohydrate content of 48.8%, a protein content of 21%, and a fat content of 3%. Rats were acclimatized for a week prior to the beginning of the experiment. This experimental study was conducted in the animal house of the college of Medicine, Ismailia, Egypt (2020). All experimental procedures were performed according to the Declaration of Helsinki and were approved by the Research Ethics Committee of the Faculty of Medicine, Suez Canal University research number (4234–2020).

2.2. Materials

The nanoparticles were prepared in the Center of Excellence, Faculty of Medicine, Suez Canal University by reduction technique as per standard protocol (Sigma Aldrich Co., Saint Louis, MO 63103, USA). 30 ml of 0.002 M of sodium borohydride (NaBH₄) was added to an Erlenmeyer flask. The solution was freshly prepared. The solution was prepared by a chemical reduction technique using sodium borohydride (NaBH₄). Polyvinyl pyrrolidone (PVP) was used to prevent agglomeration of Ag NPs as per previously published publications (El Mahdy et al., 2015; Wang et al., 2005; Zhang and Liu, 2020). Particles were prepared under optimal physicochemical circumstances including temperature and speed of stirring, according to the standard protocol. A magnetic stir bar was used with the placement of an ice bath on the stir plate. Stirring on ice was to decrease the rate of decomposition during the preparation. 0.07gm of PVP was weighed and added to borohydride solution. 2 ml of 0.001 silver nitrate (AgNO₃) was dropped into the stirring NaBH₄ solution (1 drop/sec). After adding all the AgNO₃ stirring was stopped. The resultant solution has concentration of 200 ppm of Ag NP was diluted with distilled water until a concentration of 25 ppm was obtained (1 ml of Ag NP solution to 7 ml of distilled water).

Alanine aminotransferase (ALT), Aspartate aminotransferase (AST), and Hydrogen peroxide (H₂O₂) kits were obtained (Biodiagnostics Co.; Cairo, Egypt).

2.3. Characterization of silver nanoparticles

2.3.1. Transmission electron microscopic (TEM) analysis

Characterization of Ag NPs was done via TEM in a specialized laboratory, Egypt. to evaluate the diameter and morphological properties of Ag NPs. TEM analysis revealed a sphere-shaped Ag
NPs particles that showed a limited size distribution range from 12.81 to 23.83 nm (Average: 16.5 nm) as shown in Fig. 1a.

2.3.2. Ultraviolet–Visible (UV–Vis) Spectroscopy measurements analysis

The (UV–Vis) analysis was utilized to evaluate the diameter and morphological properties of Ag NPs. The peak absorbance wavelength was at 412 nm and the particles were aggregated typically in the wavelength range of 600–800. In the absorbance wavelength 400–500, there were absence of stable particle aggregation (Cho et al., 2018). The limited distribution of the wavelength demonstrates the homogeneity of nanoparticle size; Fig. 1b.

2.4. Experiment design

At the beginning of the experiment, 18 rats were assigned randomly into three groups (n = 6/group). The groups were labeled as control group (group 1): 0.5 ml saline via oral gavage, Ag NPs groups (Group 2 & group 3): Ag NPs solution (0.5 ml of a solution containing 25 ppm Ag NPs (equal to a dose of 5 mg/kg body weight of silver). Oral gavage was given on a single daily basis. To avoid unnecessary harm or stress to the rats, oral gavage was performed using metal needles for feeding (Gauge: 16G; length: 3 in., curved) by a qualified expert (Yousof et al., 2020). On day 14, Ag NPs were stopped for both the Ag Nps groups (groups 2 & 3). To assess the direct hepatotoxic effect of Ag NPs, six animals (group 2) were anesthetized by intraperitoneal injection of thiopental Na (40 mg/kg) (Nikolić et al., 2017) and euthanized by cervical dislocation. Regarding group 3; rats were left under observation for another 14 days on the usual chow diet and free access to water to assess the presence of persistent hepatic damage following the cessation of the Ag NPs. On day 28 the same sacrificing procedures were done with groups 1 and 3. Blood samples were collected in each sacrifice day via a cardiac puncture for further analysis of liver transaminases; alanine transaminase (ALT) & aspartate transaminase (AST). Part of rat liver (10–20 g) was frozen for further homogenization to assess hydrogen peroxide (H2O2) level as an oxidative stress marker. The rest of rat livers were fixed into 10% formalin-fixed, paraffin-embedded blocks for Bdnf gene expression, histo-pathology, and immunohistochemistry.

2.5. Assessment of liver enzymes

The blood samples were let for clotting (30 min) and then centrifuged at (3000 rpm/15 min). The collected sera were then stored (−20 °C) for further liver enzymes assessment. To determine the level of liver enzymes UV/VIS Spectrophotometer (Unico S2100 Series, USA) were used. ALT and AST were assessed via colorimetric kinetic technique (Murray, 1984a, 1984b).

2.6. Assessment of hepatic hydrogen peroxide (H2O2)

The liver tissues were homogenized as per the manufacturer's guidelines H2O2 was determined by colorimetric according to (Aebi, 1984) method via UV/VIS Spectrophotometer (Unico S2100 Series, USA).

2.7. Bdnf gene expression by Real-Time PCR

The RNeasy FFPE Kit (Germany, cat. no. 73504) was used for total RNA extraction from formalin-fixed, paraffin-embedded liver tissue based on the manufacturer's guidelines (Kokkat et al., 2013). The determination of the content and purity of RNA and cDNA was examined through a Nanodrop spectrophotometer (Wilmington, DE, USA). A high-capacity cDNA Reverse Transcriptase Package (USA, cat. no. Archive) and a thermocycler (BIOMETRA® LA, USA) were used for reverse transcription. The expression of the Bdnf and B-actin genes were determined using an Applied Biosystems Step One TM Real-Time PCR apparatus. Primer assays specific for genes (willowfort, UK) were engineered and examined utilizing web Bioinformatics tools; Details are provided in the supplemental file in Table Is & Table IIs. Normalization of every target gene was done (endogenous b-actin). The 2–ΔΔCT method was utilized for the relative expression of genes (Livak and Schmittgen, 2001).

2.8. Histopathological & immunohistochemical assessments

Neutral formalin (10%) was used for the fixation of specimens for 24 h at room temperature (23–28 °C) followed by preparation of paraffin sections (5-μm-thick). The obtained sections were stained using Hematoxylin & Eosin stain (H & E) (Clark and...
Biological Stain Commission, 1981) for a general architecture of the liver, Masson’s trichrome stain for collagen fibers (Masson and Masson, 1929), periodic acid-Schiff reaction (PAS) (McMANUS, 1946) for the glycogen content in hepatocyte and immunohistochemical stain for the demonstration of apoptosis (P53, www. ABclonal.com . Catalog No.: A5761) (Anzola et al., 2004) in rat hepatocyte. Qualitative and quantitative assessments were done for histopathological and immune-histochemical changes in the liver. Quantitative measurements were done using computer software image J. Quantification of the severity of tissue damage (scoring): Regarding the liver, the severity of changes was quantitated from none (−) to severe (+++) based on the central vein congestion & dilatation; disruption of hepatic arrangement and degenerated & vacuolation of the hepatocyte.

2.9. Data analysis

IBM SPSS statistical software version 23 was applied to examine the data. Mean ± standard deviations were used to express the data, and the means were compared using ANOVA. The post-hoc LSD test was used to identify the group differences if the ANOVA was significant. Graphical abstract was presented using Mind the graph and Microsoft PowerPoint.

3. Results

3.1. Liver enzymes & H₂O₂ Assays

The results of our study revealed no significant difference between groups regarding the liver transaminases (p value greater than 0.05); reflecting no increase in liver enzymes in both experimental groups (groups 2 & 3). Assessment of hydrogen peroxide showed a statistically significant elevation in group two in comparison to the control group. Meanwhile, there was no statistically significant elevation of H₂O₂ level in group 3 in comparison to the control. Comparison between groups 2 and 3 revealed no significant differences; Table 1.

The table reveals no increase in liver enzymes in both experimental groups. Assessment of hydrogen peroxide revealed a statistically significant elevation in group 2 compared to control.

3.2. Bdnf gene Expression

Hepatic Bdnf gene expression was calculated as fold change (the control group was considered as one-fold change). Group two showed an insignificant increase in Bdnf gene expression (mean ± SD: 13.05 ± 16.81; P-value: ≥ 0.05). Group 3 in which Ag NPs were stopped for 2 weeks, showed a further but insignificant increase in the Bdnf expression (mean ± SD: 25.24 ± 39.37; P-value: ≥ 0.05).

3.3. Histopathological & Immunohistochemistry Results

A qualitative assessment of the histopathological changes in the liver using H & E stain was done. It revealed ballooning of the hepatocyte, vacuolated eosinophilic cytoplasm, and nuclear pyknosis.

Table 1

| Assessment of Liver Transaminases (U/L) and H₂O₂ (Mm/L) (n = 18, 6/group): |
|-------------------|-------------------|
| Group 1           | ALT (Mean ± SD)   | AST (Mean ± SD)   |
|                   | 12.6 ± 2.07       | 13 ± 1.7          |
|                   | 0.04 ± 0.02       |                   |
| Group 2           | 13.1 ± 1.4        | 14 ± 2.5          |
|                   | 0.36 ± 0.02*      |                   |
| Group 3           | 12.1 ± 1.1        | 12 ± 3.2          |
|                   | 0.13 ± 0.09       |                   |

* Comparison to control reveals statistical significance (p < 0.05).

4. Discussion

The goal of this research was to assess the potential effect of daily low Ag NPs dose administration on liver structure and function for a short-term period and the persistence of this harm after removal of the nanoparticles. Subacute exposure to Ag NPs can cause persistent hepatotoxicity, according to the current study. This hepatotoxicity impedes the liver’s typical quick regenerating capacity and may even postpone it once the nanoparticles are removed.

4.1. The impact of nanoparticles characteristics on their toxicity

The morphological features of nanoparticles in general and particularly Ag NPs have a great effect on the research outcomes. This is due to the variability in the chemical reactions that they induce inside the biological media of the body, as well as the liability of small-sized particles to induce more toxicity due to their ability to find a portal to the cell (Ferdous and Nemmar, 2020; Yildirim et al., 2011). Previous studies have provided different oral doses of Ag NPs for a variable time, documenting that the buildup of Ag NPs in different body tissues is dose-dependent (Kim et al., 2010, 2008). The concentration of 25 ppm (used in the current study) was reported to be the most toxic to the cell line. It is still undetermined whether the shape itself could have a role in the toxicological process or it could be a multifactorial issue (Akter et al., 2018). Another important aspect that contributes to the cytotoxic activity of AgNPs is their large surface area, which releases Ag+ ions. Because of their larger surface area to volume ratio, smaller AgNPs have a faster rate of silver ion (Ag+) dissolution in the surrounding microenvironment, resulting in improved bioavailability, enhanced distribution, and toxicity of Ag when compared to bigger NPs (Ferdous and Nemmar, 2020).

4.2. Increased H₂O₂ level in the liver

In this experiment, we detected a statistically marked increase in H₂O₂ levels in the group of animal sacrificed on day 14. This indicates an enhancement of redox activity that overcomes the cellular antioxidant capacity to scavenge the increased ROS. The increase in H₂O₂ persisted after stopping the nanoparticles in the group sacrificed on day 28. This may indicate the continued inability of cellular antioxidant activity to cope with the increased redox
Fig. 2. Photomicrograph sections in the hepatic tissue of rats using H&E stain. Fig. 2a: A liver tissue section from a control rat showing normal central veins and hepatocyte (H) which are arranged into hepatic cords radiating from the central vein (CV). Fig. 2b: Liver tissue from a rat treated with nanosilver (group 2) showing congested & dilated central vein (CV) and vacuolated ballooned hepatocytes (H) with karyolytic (K) and pyknotic (P) nuclei. Fig. 2c: Hepatic tissue from a rat in group 3 showing congested & dilated central vein (CV) and vacuolated ballooned hepatocytes (H) with karyolytic (K) and pyknotic (P) nuclei. [H&E X400].

Fig. 3. Photomicrograph sections in the hepatic tissue of rats using PAS stain. Fig. 3a: Liver tissue section from a control rat showing positive PAS reaction in the cell cytoplasm. Normal accumulations of reddish stained glycogen particles at one pole of the hepatocyte are shown. Fig. 3b: Liver tissue from a rat administered nanosilver (group 2) revealing decreased PAS reaction in the cytoplasm of hepatocyte more than the control group. Fig. 3c: Hepatic tissue section from a rat in group 3 displaying decreased PAS reaction in the cytoplasm of hepatocyte more than group 2. [PAS X400].

Fig. 4. Photomicrograph sections in the hepatic tissue of rats using MTS stain. Fig. 4a: Liver tissue section from a control rat with minimal greenish collagens around the central vein. Fig. 4b: A section from hepatic tissue from a rat given nanosilver (group 2) with no change in the greenish collagenous fibers surrounding the central vein. Fig. 4c: A section of liver from a rat from group 3 displaying no change in the greenish collagenous fibers encircling the central vein [Masson’s trichrome stain X 400].

Fig. 5. Photomicrograph sections in the hepatic tissue of rats using P53 stain. Fig. 5a: Liver tissue section from a control rat displaying mild brownish reaction. Fig. 5b: Liver tissue section from a rat given nanosilver (group 2) revealing moderate immunoreactivity of P53 in the form of brownish color. Fig. 5c: A section of hepatic tissue from a rat from group 3 with intense immunoreactivity of P53 in the form of brownish color [P53 immunostaining X 400].
state. Overall, this refers to the persistent damage that has been caused by Ag NPs. An experiment by Patlolla et al. has reported that short-term administration of high doses of silver nanoparticles (50 and 100 mg/kg) activated ROS, induced liver injury and elevated liver enzymes (Patlolla et al., 2015). Recent research reported that Ag NPs enhanced apoptosis of HepG-2 cells in human hepatocellular carcinoma in in a dosages-dependent way. They demonstrated this by activating the caspase-3 pathway by generating reactive oxygen species (ROS) (Zhu et al., 2016). A prior study found that excessive levels of H2O2 or a failure to eliminate it caused cell necrosis and death (Rosa et al., 2006). The findings of these studies, taken together, support the findings of the current investigation in terms of the mechanistic role of ROS in the production of hepatic cell injury and the augmentation of cell death following the administration of Ag NPs.

### 4.3. The immune-histopathological features of subacute toxicity by Ag NPs

PS3 is a tumor suppressor gene. When PS3 is moderately stimulated within physiological limits, it exerts a beneficial function. The physiological apoptosis gets rid of the damaged or aged cells without releasing pro-inflammatory cytokines and eliciting an immune response. In contrast, hypo-activation or hyper-activation can lead to liver pathologies that can ultimately lead to HCC (Krstic et al., 2018; Link and Iwakuma, 2017). This is due to the improper removal of cells with mutated genes. In the same consensus, excessive or sustained apoptosis can lead to acute or chronic sustained inflammations and hepatic diseases (Cao et al., 2016; Guicciardi et al., 2013). The findings of the present study showed a toxic effect of Ag NPs on the liver at the structural level. Immuno-histochemical assessment of the apoptotic gene PS3 revealed increased expression in the experimental group administered Ag NPs for 14 days. The present study has a significant finding which is the aggravated increase in PS3-labeled hepatocytes in the group in which Ag NPs has stopped for fourteen days. The previous finding could reflect the persistent effect of Ag NPs after their withdrawal. A previous study could support our findings, albeit the different protocols and the type of silver nanoparticles. In that study, the researchers administrated a single dose (20% (w/v) aqueous solution – 7.9 ± 0.95 nm in size) of citrated Ag NPs into rabbits. They reported irreversible damage in liver cells that lasted for at least a month after a single dosage of citrated Ag NPs cytoplasmicity (Kim et al., 2019). The main concern in this argument is the liability of this pathologic apoptosis to induce fibrosis or hepatic cancer (Guicciardi and Gores, 2005) which is an issue that should be thoroughly investigated due to the widespread introduction of Ag NPs in many aspects of daily life. In our work, the histological alterations in the liver caused by Ag NPs were assessed qualitatively and quantitatively, proving their toxicity. A healthy liver’s regenerative capacity is slow. When the liver is injured, on the other hand, this capability is increased to compensate for the cell loss (Krstic et al., 2018). Based on what has been published in earlier studies, there are two types of regeneration models of the liver tissue (Tao et al., 2017). The first model occurs in a unique pattern and follows partial hepatectomy (removal of 2/3 of liver tissue), and it takes about a week for the hepatocyte to regenerate and compensate for the loss. The other model is what occurs in liver insult via toxins or viruses. The oval stem cells operate and develop into hepatocytes and biliary cells to heal the substantial damage to the liver cells in this later model. Others have found disagreements over whether regenerated liver cells come from mature hepatocytes or a specific cell type (Gilgenkrantz and l’Hortet, 2018; Moreno-Marín et al., 2017). The high regenerative capacity of the liver can start as early as 24 to 72 h after drug-induced liver damage and can be seen along with the signs of liver injury on microscopic examination (Michalopoulos, 2007; Steup et al., 1993). As a result, we expected the disease to recover in two weeks, with the regenerative features overcoming the indicators of liver injury. This, however, was not the case. Many causes attribute to this unexpected outcome. To begin with, even when the nanoparticles are removed, a portion of them can precipitate in the liver and so cannot be entirely eliminated. It is uncertain if this damage can be reversed by the liver’s long-term regenerative power or if it will result in chronic liver damage that has yet to be determined. Another explanation could be the ability of some noxious materials to induce long-lasting damage even in a single given dose. Ag NPs could be one of these material categories that induce persistent damage over short-term use (Kim et al., 2019). Ag NPs have also been shown to suppress the growth of pathogens like Herpes, HIV, and Hepatitis B, according to studies. The presence of the Ag + ion and its capacity to aggressively attach to the negatively charged proteins of DNA and RNA, preventing cell replication, is thought to be the cause of lower cellular proliferation caused by Ag NPs applications. This, probably, could indicate that a similar reaction could occur in somatic liver cells (Nayek et al., 2021).

### 4.4. The role of hepatic Bdnf gene in Ag NPs toxicity

BDNF regulates various aspects of brain activity and is crucial for maintaining central energy balance. It is found in organs in
the periphery that are incorporated in fat and glucose metabolism, like the liver, although its significance in these tissues is uncertain. The findings suggest that through decreasing Peroxisome proliferator-activated receptor alpha and fibroblast growth factor 21, hepatic BDNF may aid the establishment of hepatic illnesses in response to a high fat diet (Teillon et al., 2010). A previous study found that the serum BDNF level is significantly lowered in patients with hepatitis B virus-induced liver cirrhosis (Shu et al., 2019). Another study has correlated the Ag NPs to the suppression of BDNF-stimulated cell survival in neuroblastoma cells treated with Ag NPs (Park et al., 2017). In the current study, although BDNF expression was lower in liver tissue after Ag NPs cessation, this difference was not statistically significant. BDNF and its receptor were only found in hepatic carcinoma tumor tissue and cell lines in a previous investigation, though not in nontumorous hepatocytes or normal cell lines. Therefore, elevated BDNF expression may promote tumor growth (Roesler et al., 2011; Yang et al., 2005). This may point to that although there were structural damage and apoptosis after Ag NPs withdrawal, the cells did not pass to a tumorigenic transformation, may be due to the short time of nanoparticle exposure.

4.5. Liver enzymes in Ag NPs subacute toxicity

Liver enzymes, ALT & AST, showed insignificant changes between the control and experimental groups. Although we did not expect that due to apparent cellular affection at the histological level, other studies concerned with the liver insult by diseases have shown similar results. Previous studies investigated the increase in the pro-apoptotic gene P53 in non-alcoholic liver steatosis. It has related the elevated transferases to the stage of the steatosis and indicated the insignificant correlation between liver transferases elevation and the expression of P53 (Panasiuk et al., 2006). Researchers found that when 3 groups of adult male Wister rats were given an intraperitoneal injection of Ag NPs at concentrations of 5, 10, and 100 ppm for 7 days, serum Glutamic Oxaloacetic Transaminase (SGOT) and Serum Glutamic-pyruvic Transaminase (SGPT) were insignificantly elevated at different time points compared to the control group. Nevertheless, they have reported partial damage to the liver tissue on histopathological examination (Monir Doudi, Mahbubeh Setorki, 2014). Another study that examined the effect of Ag NPs on the liver at varying doses for 28 days, documented no considerable effect of silver nanoparticles on liver transferases, although it also reported up-regulating activity of liver caspase-3 (Mahsa et al., 2016). In contrast, a prior study on silver nanoparticle toxicity found that the experimental groups had higher levels of liver enzymes than the control group (Heydnejad et al., 2015). Other researchers documented a dose-dependent (concentrations: 5, 10, 20, and 40 ppm of Ag NPs) elevation of liver enzymes although they did not detect histopathologic changes in the liver (Shahin et al., 2014). Therefore, the structural heterogeneity of silver nanoparticles, the route of delivery, and the duration of treatment could all be factors in the disparities among experiments (Ajdary et al., 2018; Mao et al., 2018).

5. Conclusion

In general, the current study found that subacute Ag NP exposure can have long-term hepatotoxic consequences. This hepatotoxicity impedes the liver’s typical quick regenerating capacity and may even postpone it when the nanoparticles are eliminated. However, long-term follow-up studies for various periods of time after terminating Ag NPs treatment are still needed to confirm these findings. Moreover, the introduction of variable doses to validate the effect of time in relation to the dosage on liver toxicity and regenerative capacity should be considered. The role of BDNF in subacute hepatic toxicity by Ag NPs particles and its interaction with the P53 needs to be elucidated. As a consequence, these findings raise concerns about the uncontrolled and unmonitored massive use of Ag NPs in recent decades. Normal levels of liver enzymes, if not paired with evident symptoms indicating hepatic disease, might be deceiving and postpone medical attention, which can lead to future health dangers as a result of the cumulative effect of using these nanoparticles.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Author contribution

SHY: generating the idea of the research, methodology formulation and follow up of laboratory experiments, searching the methods, editing of the manuscript and discussing the results. HE: editing & performing methodology of histopathologic and immune-histochemistry assessment, and performing statistical analysis and discussing & editing of their results. MH: Molecular Biology analysis, shared in methodology and shared in result analysis and presentation. KEI-S: generating the idea of the research and methodology formulation and follow up of laboratory experiments, partial share in editing of the manuscript and discussing the results, ShA editing the paper, editing the methodology and conceptualization. All authors shared in revising the manuscript, performing the statistical analysis and results presentation. The first and the last authors provided similar contributions in authorship.

Appendix A. Supplementary material

Supplementary data to this article can be found online at https://doi.org/10.1016/j.sjbs.2022.02.054.

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