Effect of cooling rate and composition on microstructure and mechanical properties of ultrahigh-strength steels

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Abstract
The influence of cooling rate on the microstructure and mechanical properties of two new ultrahigh-strength steels (UHSSs) with different levels of C, Cr and Ni has been evaluated for the as-cooled and untempered condition. One UHSS had higher contents of C and Cr, while the other one had a higher Ni content. On the basis of dilatation curves, microstructures, macrohardness and microhardness, continuous cooling transformation diagrams were constructed as a guide to heat treatment possibilities. Cooling rates (CRs) of 60, 1 and 0.01 °C/s were selected for more detailed investigations. Microstructural characterization was made by laser scanning confocal microscopy, field emission scanning electron microscopy combined with electron backscatter diffraction, electron probe microanalysis and X-ray diffraction. Mechanical properties were characterized using macrohardness, tensile and Charpy V-notch impact tests. UHSS with the higher C and Cr contents showed lower transformation temperatures and slower bainite formation kinetics than that with the higher Ni content. Higher cooling rates led to lower volume fractions and carbon contents of retained austenite together with finer prior austenite grain size, as well as effective final grain size and lath size. These changes were accompanied by higher yield and tensile strengths. The best combinations of strength and toughness were obtained with martensitic microstructures and by avoiding the formation of granular bainite accompanied by proeutectoid carbides at low CR. For the cooling rates studied, UHSS with the higher C and Cr contents showed the higher hardness and strength but at the cost of toughness.

Keywords Ultrahigh-strength steel · Electroslag remelting · Continuous cooling transformation diagram · Microstructure feature · Mechanical property

1 Introduction
Nowadays, ultrahigh-strength steels (UHSSs) are widely used in many applications such as automotive, locomotive and truck components, pressure vessels, offshore platforms, engineering machinery, the mining, military and aerospace industry, due to their good combinations of strength, toughness and ductility [1–5].

Recently, different grades of steels have been developed as potential replacements for the high-cost ultrahigh-strength steels containing high levels of Co and Ni such as AerMet100, HP 9-4-20, HP 9-4-30 and AF1410 to attain substantially reduced costs and processing time. Based on a medium-carbon V-microalloyed composition alloyed with moderate levels of Si, Ni, Cr, Mo and W, Dilmore and Ruhlman [6] developed the ultrahigh-strength Eglin steel that, after tempering at 260 °C, achieves a hardness of 458 HV, an ultimate tensile strength of 1685 MPa, a yield strength of 1392 MPa, elongation to fracture of 17.5% and a Charpy V-notch impact toughness of 37 J at room temperature. With the aim of improving the strength–toughness balance achievable with Eglin steel, especially in thicker sections with low cooling rates (CRs), Vartanov [7] designed a medium-carbon steel microalloyed with V in combination with Nb or Ti but less Ni and W than that in
the Eglin steel. After tempering at 216 °C, the designed steel obtains a hardness of 594 HV, an ultimate tensile strength of 2055 MPa, a yield strength of 1578 MPa, an elongation to fracture of 13.5% and a room-temperature Charpy V-notch impact toughness of 35 J.

The main difference between the currently investigated steels is in C, Cr and Ni contents. C is the most important element used to increase hardness and strength through (i) lowering of the phase transformation temperatures during continuous cooling, thereby refining the microstructure and increasing the dislocation density, (ii) an increase in solid solution strengthening, and (iii) carbide precipitation strengthening [8]. However, the level of carbon alloying needs to be limited to avoid detrimental effects on weldability and toughness. Cr is used to enhance strength and hardenability through transformation strengthening and, when it is combined with C, it is a powerful hardening alloy element. Also, it increases corrosion resistance. Ni has a positive influence on toughness and also helps to stabilize retained austenite (RA).

The present work has been undertaken in order to extend the studies of Dilmore and Ruhlman [6] by exploring the possibilities of producing ultrahigh-strength tough steel with even lower cost via the use of recycled steels, ferroalloys and nickel alloys through air melting and electroslag remelting applied to two low-purity Eglin steels with traces of Ti and Nb as described in previous publication [9]. The use of simpler liquid steel treatment means that the impurity and inclusion contents are higher than those obtained with vacuum treatments [9, 10]. In the present paper, we explore the phase transformations, together with the resultant microstructure and mechanical properties of the current lower-purity Eglin steels with traces of Ti and Nb for the wide range of cooling rates that can be encountered in practical processing. Mechanical properties are investigated using tensile, hardness and Charpy V-notch (CVN) impact testing. Microstructures are characterized with regard to the fractions of martensite, bainite and retained austenite together with the sizes of prior austenite grains and the sizes of grains defined by low-angle and high-angle grain boundaries.

2 Experimental

Two experimental heats with the chemical compositions given in Table 1 were designed, melted using an induction furnace and cast into ingots of 70 mm in diameter in the Steel Technology Department, Central Metallurgical Research and Development Institute (CMRDI), Egypt. For each steel, after discarding 30 mm from the bottom, the cast ingots were cut into two equal lengths and they were subsequently reheated to 1100 °C and held for 1 h followed by forging in the temperature range of 950–1100 °C to a bar with a cross section of 28 mm × 30 mm, which was then air-cooled with an average cooling rate of 0.3 °C/s over the temperature range of 950–35 °C. For each steel, one of the forged bars was used as a consumable electrode for electroslag remelting (ESR) under a synthetic fused slag containing about 70% CaF₂, 15% Al₂O₃ and 15% CaO in weight percentage. The 70-mm-diameter ESR ingot was subsequently forged and cooled using the same parameters that were used for the base ingot from the induction furnace.

Continuous cooling transformation (CCT) diagrams of the investigated steels were produced using dilatation experiments on a Gleeble 3800 thermomechanical simulator with cylindrical samples of 6 mm in diameter and length of 20 mm machined with the cylindrical axis parallel to the forging direction. The critical transformation temperatures A₁, A₃ and A₅ were determined using the tangent method on dilatation diagrams produced during heating at 5 °C/s. The transformation temperatures on cooling were determined using the tangent method on the dilatation curves obtained after heating the samples at 5 °C/s to 1100 °C, holding for 3 min, cooling to 950 °C at 2 °C/s, holding for 3 min and finally cooling to room temperature at linear cooling rates ranging from 0.01 to 60 °C/s as illustrated in Fig. 1.

A laser scanning confocal microscope (LSCM) was used to investigate all samples after phase transformation. Samples were cut, mounted, ground, polished down to 1 μm using a diamond suspension and then ultrasonically cleaned. All samples were etched using 4 vol.% picric acid for 30–60 s followed by fresh 2 vol.% Nital solution for 30–60 s. After each step, samples were washed with water and ethanol and then dried in air. In order to identify martensitic and bainitic regions for the determination of microhardness, tinfoil etching was used. The samples were first etched using 4 vol.% picric acid for 1 min and then etched for 15–20 s using a 10% aqueous solution of sodium metabisulfite (10 g of Na₂S₂O₃ in 100 mL distilled water). The samples were oscillated strongly during the whole etching process. After each step, the surface of the sample was cleaned using water and ethanol followed by drying [11].

Microhardness tests were performed on a CSM microindentation tester using a maximum load of 0.25 N, a loading rate of 0.50 N/min, an unloading rate of 0.50 N/min and a pause time of 10 s. At least 15 random measurements across the samples were used to determine characteristic average values. CCT diagrams were constructed based on the results from dilatation curves, final microstructures and microhardness values. HV10 macrohardness measurements were made using a Duramin-A300 (Struers) macrohardness tester at 10 random positions...
The microstructures of the samples used for the microstructure investigations after repolishing.

The thermal cycles in Fig. 1 with cooling rates of 60, 1 and 0.01 °C/s were selected for more detailed studies of the effect of the cooling rate on the microstructures and mechanical properties. Using the Gleeble thermomechanical simulator, samples with the dimensions of φ6 mm × 90 mm were thermally cycled on using a free span of 20 mm which gave a uniform microstructure and hardness over a distance of 15 mm in the middle of the free span. Samples were subsequently machined to round tensile samples with a diameter of (4 ± 0.1) mm and a gauge length of (16 ± 0.1) mm according to ASTM standard E8. Tensile testing was done using a Zwick/Roell machine with a maximum load capacity of 100 kN. CVN impact test samples were thermally cycled using the Gleeble thermomechanical simulator with free span of 5 mm which gave uniform microstructure and hardness over 5 mm in the middle of the samples, and then the notch was machined in the middle of the samples at the location of the control thermocouple. Impact tests were performed at room temperature (20 °C) and −40 °C according to ASTM E 23-05 on standard sub-sized CVN impact samples (5 mm × 10 mm × 55 mm) using a calibrated Charpy impact test hammer with a capacity of 350 J. The results are average values of three samples. The thermal cycle for the tensile and impact samples with a cooling rate of 0.01 °C/s was performed in a controllable furnace and left in the furnace for cooling with an average cooling rate of 0.01 °C/s.

The microstructures of the samples used for the macrohardness tests were also characterized using a field emission scanning electron microscope (FESEM, Zeiss Sigma). Average effective grain size and grain boundary misorientation distributions were measured using an energy-dispersive X-ray (EDAX) electron backscatter diffraction (EBSD) system on FESEM with an accelerating voltage of 15 kV, a magnification of 2000 and a step size of 0.15 μm. Grain boundaries with misorientation higher than 2° and higher than 15° were considered as lath boundaries and high-angle grain boundaries, respectively [12, 13]. High-angle grain boundaries refer to the boundaries of blocks or packets [14, 15]. Equivalent circular diameter (ECD) was used to define the lath size and the effective grain size as defined by the high-angle grain boundaries. For EBSD examination, the sample surface was first polished with a diamond suspension down to 1 μm and then chemically polished using a 0.05-μm colloidal silica suspension. To determine the prior austenite grain size, different EBSD runs were performed at low magnification with an accelerating voltage of 15 kV, a magnification of 500 and a step size of 0.65 μm. To reveal prior austenite grain (PAG) structure, EBSD orientation data were used to reconstruct the prior austenite using MATLAB software combined with the MTEX texture and crystallographic analysis toolbox [16–18]. The linear intercept method was then used to determine prior austenite grain size (PAGS) from the reconstructed image.

The volume fraction of retained austenite was determined using a Rigaku SmartLab X-ray diffractometer with Co Kα radiation under the following conditions: accelerating voltage of 40 kV, current of 135 mA, scanning speed of 1.0039 (°)/min, step size of 0.05° and range of 40° > 20 > 130°. Samples from the fractured sub-sized CVN impact samples with the dimensions of 10 mm × 10 mm × 5 mm were cut, ground, polished using a diamond suspension down to 1 μm and then ultrasonically cleaned in ethanol to remove any dust or particles on the surface. To determine the volume fractions of austenite, X-ray diffraction data were treated using the whole profile Rietveld refinement analysis using PDFXL2 analysis software. The carbon content of the retained austenite (C_r) was estimated from the lattice parameter, i.e., $a = 0.3578 + 0.0033C_r$ [19], where $a$ is the lattice parameter of the retained austenite, nm.

To determine the chemical constituents in grain boundary carbides, Jeol JXA-8200 electron probe microanalysis (EPMA) using line scan analyses with a step size of 0.4 μm was made in the case of samples cooled at 0.01 °C/s. Samples for EPMA were prepared by repolishing the microstructural samples and lightly etching using 4 vol.% picric acid for 20 s followed by fresh 2 vol.%
Nital solution for 20 s. The results from EPMA were confirmed using FESEM energy-dispersive X-ray spectroscope (EDS) analysis of the carbides formed at the PAG boundaries using the same samples.

3 Results and discussion

3.1 CCT diagrams

The values of $A_{c1}$ and $A_{c3}$ for UHSS A are 746 and 831 °C, and those for UHSS B are 734 and 845 °C, respectively. CCT diagrams based on dilatation curves, LSCM micrographs, macrohardness and microhardness values are shown in Fig. 2. $M_s$ and $M_f$ are the martensite start and finish transformation temperatures. $B_s$ and $B_f$ are the bainite start and finish transformation temperatures. The effect of C content and Cr/Ni ratio is clearly observed from Fig. 2: $A_{c1}$ increased as a result of increasing Cr/Ni ratio and a small increase in C content, while $A_{c3}$, $M_s$ and $B_s$ decreased. Typical LSCM micrographs after phase transformation at different cooling rates are shown in Figs. 3 and 4. The higher hardness values of UHSS A (635-416 HV) compared to those of UHSS B (580-393 HV) result from the higher hardenability provided by the higher C content and Cr/Ni ratio. This results in lower transformation temperatures and thereby finer microstructures and higher dislocation densities. The microhardness values of the transformation products are given in Fig. 5. In UHSS A, martensite with hardness ranging from 635 to 570 HV was obtained over the wide range of cooling rates from 60 to 0.2 °C/s and a mixture microstructure of martensite and bainite was obtained from 0.08 to 0.02 °C/s with hardness range from 543 to 430 HV. However, the main transformation product at 0.01 °C/s with hardness of 416 HV was granular bainite. Also, carbides were formed at the prior austenite grain boundaries as shown in Fig. 3j. The same transformation products were observed in UHSS B but with some difference as bainite started to form at the cooling rate of 0.2 °C/s instead of 0.08 °C/s in UHSS A due to its lower C content and Cr/Ni ratio, which decrease the hardenability and shift the bainite transformation curve to a higher cooling rate.

In UHSS A, the microhardness of the martensite increased from the slowest cooling rates up to 15 °C/s after which the hardness is constant, see Fig. 5a. This behavior may be due to the increasing loss of carbon from the martensite laths into an increasing volume fraction of retained austenite as explained below [20]. Increasing the cooling rate from 0.01 to 0.08 °C/s has no effect on the microhardness of the bainite as shown in Fig. 5a. The same trend of the microhardness of the martensite was observed for UHSS B (Fig. 5b). However, the microhardness of the bainite increased from 0.01 to 0.08 °C/s and showed no further increase after this cooling rate. The clear difference in the microhardness of bainite over CR range from 0.01 to 0.08 °C/s between the investigated steels can be explained by the formation of a large volume fraction of proeutectoid carbides at the slowest cooling rate in the case of UHSS A. This leads to an increasing consumption of C and other alloying elements like W, Mo and Cr with decreasing CRs, which, in turn, leads to a drop in the hardness of the bainite. The effect can also be seen in the hardness of the martensite, too. The lower incidence of proeutectoid carbide in UHSS B seems to make this effect less apparent.

3.2 Microstructure of UHSSs A and B

SEM micrographs of UHSSs A and B with cooling rates of 60, 1 and 0.01 °C/s are shown in Fig. 6. They confirm the observations from LSCM and microhardness testing that the microstructure mainly consists of martensite at the cooling rates of 60 and 1 °C/s and mainly granular bainite at the cooling rate of 0.01 °C/s. The large increase in PAGS as shown below and the slowest cooling rate enhance opportunity to form granular bainite which is characterized by the lack of carbides as shown in Fig. 6c, f [21]. Figure 7 gives the results of EPMA line analysis across the grain boundaries in the samples with CR of 0.01 °C/s. Enhanced levels of Cr, W and Mo in UHSS A (Fig. 7c) and W in UHSS B (Fig. 7d) are detected. FESEM-EDS analyses shown in Fig. 8 reveal that these enhanced levels of Cr, W and Mo are related to the formation of proeutectoid alloy-rich carbides at PAG boundaries as a result of the slow cooling rate. In the case of UHSS A, two types of proeutectoid carbides are observed: complex carbides rich in Fe, W and Mo and iron carbides containing some Cr. In UHSS B, carbides at PAG
boundaries are much less common than those in UHSS A as can be seen by comparing with Figs. 3j, 4j, and 8a, b. It is presumably due to the lower C and Cr contents of UHSS B. Some carbides are observed to mainly contain iron with only low levels of Cr, W and Mo.

Support for the conclusion that the carbides are proeutectoid in nature is provided by thermodynamic calculations using the commercial software Thermo-Calc together with TCFE7 database shown in Fig. 9. Table 2 shows the predicted volume fraction of carbides in equilibrium with austenite at 755 °C. It can be seen that the compositional difference between the steels, especially their C and Cr contents, results in clear difference in the predicted equilibrium volume fractions and composition of the carbides. Even if equilibrium compositions and volume fractions were not achieved during cooling at 0.01 °C/s, the relative volume fractions shown in Table 2 are in line with the observation that UHSS A shows a clearly higher incidence of proeutectoid carbides than UHSS B.

Figure 10 shows combined EBSD inverse pole figures (IPF) and image quality (IQ) maps of the investigated steels with different cooling rates. Figure 11 shows grain boundary misorientation distributions, lath sizes, effective grain sizes and effective grain sizes at 90% in the cumulative grain size distribution for the investigated steels at different cooling rates. Table 3 summarizes the reconstructed PAGS, volume fractions of retained austenite and their carbon contents.
Fig. 4 LSCM micrograph of UHSS B with cooling rates of 60 (a), 30 (b), 15 (c), 5 (d), 1 (e), 0.6 (f), 0.2 (g), 0.08 (h), 0.02 (i) and 0.01 °C/s (j)

Fig. 5 Effect of cooling rate on microhardness of martensite and bainite in UHSS A (a) and UHSS B (b)
In the case of UHSS A, as shown in Figs. 10 and 11, the microstructural parameters are changed as a result of cooling rate. As can be seen from Fig. 11a, the average ECD martensitic lath sizes are increased slightly from 0.88 to 0.98 μm when the cooling rate decreased from 60 to 1 °C/s, while the effective (high-angle grain boundary) grain size increased from 1.04 to 1.23 μm. The value D90% describes the effective grain size at 90% in the

Fig. 6 SEM micrographs of UHSS A (a–c) and UHSS B (d–f) with cooling rates of 60 (a, d), 1 (b, e) and 0.01 °C/s (c, f). Plane of micrographs is parallel to forged bar axis

Fig. 7 EPMA micrograph and line analysis across grain boundaries of UHSS A (a, c) and UHSS B (b, d) with cooling rate of 0.01 °C/s

In the case of UHSS A, as shown in Figs. 10 and 11, the microstructural parameters are changed as a result of cooling rate. As can be seen from Fig. 11a, the average ECD martensitic lath sizes are increased slightly from 0.88 to 0.98 μm when the cooling rate decreased from 60 to 1 °C/s, while the effective (high-angle grain boundary) grain size increased from 1.04 to 1.23 μm. The value D90% describes the effective grain size at 90% in the
Fig. 8 Morphology and EDS analysis of carbides formed at prior austenite grain boundaries in UHSS A (a, c, e, g) and in UHSS B (b, d, f, h) at cooling rate from austenitization temperature (950 °C) of 0.01 °C/s
cumulative size distribution. This value can be calculated based on the number of grains D90% (GN) which we used in earlier publication [10] or calculated based on the cumulative grain area definition D90% (GA). Both D90% (GN) and D90% (GA) are calculated based on EBSD data for grains with misorientation angles higher than 15°. In the case of D90% (GN), all grains have the same weight, and 90% of the grains have an ECD less than D90% (GN). However, in the case of D90% (GA), each grain is weighted according to its area, and D90% (GA) is ECD below which 90% of the cumulative area of all grains is located. In other words, 10% of the sample cross section contains grains larger than D90% (GA). Now, we prefer to use D90% (GA) because it gives more weight to the larger grains, while D90% (GN) is very sensitive to the cutoff grain size used at the lower end of the grain size distribution. Both values are given in Fig. 11a, b. As can be seen from Fig. 11a, D90% (GA) increased from 4.8 to 6.9 μm when the cooling rate decreased from 60 to 1 °C/s. PAGS increased from 19.8 to 26.2 μm resulting from more extensive grain growth during the longer time at temperatures below 950 °C at the slower cooling rate [20], see Fig. 12a, b and Table 3. At the slowest cooling rate of 0.01 °C/s, the above parameters are quite different as PAGS increased significantly to 46.2 μm. The microstructure is now fully bainite with the average ECD bainitic lath size, effective grain size (EGS) and D90% (GA) equal to 1.01, 1.06 and 12.65 μm, respectively. The opposite trends observed regarding the effect of cooling rate on D90% (GN) and D90% (GA) for the two slowest cooling rates are due to the difference in the grain size distributions: Compared to CR of 1 °C/s, at 0.01 °C/s, the grain size distribution contains a larger number of small grains, which causes a reduction in D90% (GN), as well as an increase in the size of the largest grains, which leads to an increase in D90% (GA).

Changes in cooling rate have similar effects on the microstructural features of UHSS B: PAGS increased from 35.5 to 36.9 μm as CR decreased from 60 to 1 °C/s. As shown in Fig. 11b, the average ECD martensitic lath size, EGS and D90% increased from 0.75, 0.98 and 4.4 μm to 0.88, 1.08 and 5.6 μm, respectively, while these microstructural features for the bainite obtained at the

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**Table 2** Volume fraction of carbides formed with austenite at 755 °C

| Carbides | Chemical constitution | Volume fraction of carbides/vol.% |
|----------|-----------------------|----------------------------------|
| MC       | (V, W)₃ (C)         | UHSS A: 0.001  UHSS B: 0.001    |
| M₃C₆     | (Fe, Cr)₂₀ (W, Mo)₃ (C)̥ | UHSS A: 0.018  UHSS B: 0.004    |
| M₆C      | (Fe)₂ (W, Mo)₂ (Fe, W, Mo, Cr)₂ (C)₁ | UHSS A: 0.002  UHSS B: 0.006    |
| Total    |                        | UHSS A: 0.021  UHSS B: 0.012    |
Fig. 10  IPF combined with IQ map of UHSS A (a–c) and UHSS B (d–f) with CR of 60 (a, d), 1 (b, e) and 0.01 °C/s (c, f). Plane of micrographs is parallel to axis of forged bars.

Fig. 11  Average lath size, average effective grain size and effective grain size at 90% in cumulative grain size distribution (a, b) together with grain boundary misorientation distributions (c, d) for UHSS A (a, c) and UHSS B (b, d)
slowest cooling rate become 0.71, 0.92 and 15.7 μm, respectively, and PAGS is 55.0 μm. The mean EGSs are somewhat smaller in the case of the bainitic microstructure than in the martensitic microstructures. This may result from the strengthening of the austenite during transformation to bainite due to a concomitant increase in its carbon content resulting in a decrease in the bainitic block size [22]. This effect is also seen in the higher frequency of high-angle boundary peaks in bainite compared to that in martensite (see Fig. 11c, d). The effect of cooling rate on PAGS and D90% is shown in Fig. 13. In both investigated steels, as the cooling decreased, both PAGS and D90% increased sharply. Also, Fig. 13 shows that there are strong correlations between PAGS and D90%. These results are in line with the commonly observed reduction in block and packet size resulting from a refinement of PAGS [23]. The difference in PAGS (see Table 3) is probably due to the large time difference that the austenite spends at high temperatures before the phase transformations begin. For example, to decrease the temperature from 950 to 900 °C, 83, 0.83 and 0.016 min are required for CRs of 0.01, 1 and 60 °C/s, respectively. The difference in PAGS is particularly noticeable between CRs of 1 and 0.01 °C/s.

For both steels, the lath and mean effective grain sizes are only slightly affected by CR; they mainly depend on the type of microstructure formed and not on the changes in PAGS [24]. They increased slightly in the case of the fully martensitic structures when the cooling rate decreased from 60 to 1 °C/s. However, with a further decrease in the cooling rate to 0.01 °C/s, the formation of bainite led to a

| Steel   | Cooling rate/(°C s⁻¹) | PAGS/μm        | V₉₅/₀° % | Cₜ/₀° % |
|---------|-----------------------|----------------|----------|---------|
| UHSS A  | 60                    | 19.83 ± 1.40   | 4.60 ± 0.30 | 0.21    |
|         | 1                     | 26.18 ± 1.18   | 5.08 ± 0.19 | 0.41    |
|         | 0.01                  | 46.21 ± 7.57   | 17.16 ± 0.12 | 0.85    |
| UHSS B  | 60                    | 35.53 ± 3.64   | 2.30 ± 1.10 | 0.36    |
|         | 1                     | 36.89 ± 1.76   | 2.33 ± 0.11 | 0.67    |
|         | 0.01                  | 55.04 ± 4.49   | 12.25 ± 0.11 | 0.82    |

aError bars are 95% confidence intervals for mean value
bError bars are standard deviations

Fig. 12 Reconstructed PAG using EBSD results together with MTEX and MATLAB software for UHSS A (a–c) and UHSS B (d–f) with CR of 60 (a, d), 1 (b, e), and 0.01 °C/s (c, f). Plane of micrographs is parallel to axis of forged bars
decrease in their values. The current results show that there is a strong correlation between PAGS and D90% based on the cumulative grain area, which increased sharply when CR was at its lowest.

From Fig. 11a, b, it can be seen that although PAGS of UHSS A are lower than those of UHSS B, the lath and mean effective grain sizes in UHSS A are higher than those in UHSS B, which confirms the above conclusion that the lath and mean effective grain sizes are not dependent on the PAGS but depend on the chemical composition and whether bainite or martensite forms.

Figure 14 and Table 3 show that RA percentage is small and practically independent of CR in the case of the martensitic microstructures, i.e., at 60 and 1 °C/s, but appreciable amounts of RA are found when the microstructure is bainite. The formation of RA in bainite is assisted by the presence of approximately 0.6% Si in the steel, known to reduce the kinetics of cementite precipitation from austenite [25–27]. Also, it is assisted by the very slow cooling rate [28] and the presence of Ni, both of which are known to help stabilize RA [29]. Slower cooling rates give more time for carbon diffusion into and within the austenite [30]. All these factors can increase the stability of austenite and reduce its probability for full transformation. Also, the presence of the granular bainite may contribute to the stability of the retained austenite [31].

3.3 Mechanical properties of UHSS A and B

The mechanical properties of the investigated steels after the thermal cycles (shown in Fig. 1) with cooling rates of 60, 1 and 0.01 °C/s, and the temperatures for 50% transformation are presented in Table 4, Figs. 15 and 16, respectively. As commonly observed [24], the hardness, ultimate tensile strength (UTS) and yield strength (YS) increase as the transformation temperature drops with increasing CR. This trend is due to various interrelated factors such as increasing carbon content and other alloying elements in solid solution in the martensite, as well as refinement of the martensite lath and prior austenite grain sizes. These results are in good agreement with the results obtained by Kennett and Findley [32] who conclude that strength is increased with refinement of PAGS in microalloyed ASTM A514 steel in the as-quenched and low-temperature tempered state. Lan et al. [33] observed that YS and UTS of a low-carbon bainitic steel increased due to the refinement of PAGS, packets and blocks. Sun et al. [34] stated that the effective grain size (block size) is the controlling factor for YS and toughness of martensitic steel. In addition to the above parameters, lower strength and hardness result from the appearance of granular bainite with proeutectoid carbides at PAG boundaries. Also, the formation of grain boundary carbides may lead to a decrease in toughness. Increasing the cooling rates leads to an improvement of the strength, but elongation to fracture (El) is decreased because of the presence of hard martensite.

The toughness properties including impact toughness at room temperature and −40 °C and ductile fracture (DF) percentage are increased when the cooling rate increased from 0.01 to 1 °C/s and then decreased when CR increased to 60 °C/s. The same trend was observed by Kumar et al. [20]. The decrease in toughness at the highest cooling rate (60 °C/s) may be attributed to the very high strength values, and the presence of less retained austenite compared with the other cooling rates [35]. The lowest cooling rate (0.01 °C/s) shows the lowest toughness even though it is associated with the lowest values of strength and highest retained austenite contents. This is probably due to the large D90% (GA), associated with the large PAGS, the
carbides formed at the grain boundaries and the appearance of granular bainite, which has low resistance to crack propagation due to its large crystallographic packet size and crack nucleation sites with coarse martensite–austenite (M–A) constituents [21, 36]. Similar results were found by Luo et al. [24] who studied the effect of different cooling rates (using water, oil, rapid air and sand cooling) on the substructure of mixed lath martensite–bainite microstructures in NiCrMoV low-carbon steels and the effect on the toughness properties. They suggested that the presence of granular bainite at the slowest cooling rate was responsible for a loss of toughness.

Since the two steels have received identical thermomechanical and heat treatments, the higher strength and lower
toughness of UHSS A compared to UHSS B are presumably a consequence of somewhat higher contents of C and Cr and lower content of Ni giving rise to the microstructural differences discussed above. The larger effective grain size of UHSS A, often associated with a lower strength, has presumably been countered by higher contributions from solid solution and precipitation strengthening, and possibly a greater transformation induced plasticity effect stemming from a higher content of RA.

The relationship between the yield strength and impact toughness at room temperature (RT) and $-40^\circ C$ is presented in Figs. 17 and 18. It is clear from the figures that for the martensitic microstructures, there is an inverse
relationship between yield or tensile strength and CVN impact toughness. The trend line for the martensitic microstructures does not extend to the bainitic microstructures obtained at the slowest cooling rate, however. In that case, impact toughness is poorer despite much lower yield and tensile strengths. Therefore, the best toughness is obtained with the lowest cooling rate that avoids the formation of bainite. This should be taken into account when designing practical parts and their cooling strategies.

4 Conclusions

1. For the steel with the highest C and Cr contents, fully martensitic microstructures were obtained down to cooling rates of 0.08 °C/s, but the hardness of the martensite decreased substantially as the cooling rate decreased. Mixtures of bainite and martensite form at cooling rates of 0.02 °C/s or 0.08 °C/s depending on the composition. In both steels, the main transformation products formed at 0.01 °C/s were granular bainite with proeutectoid carbides formed at PAG boundaries.

2. RA percentage is small and practically independent of cooling rate in the case of martensitic microstructures, i.e., at 60 and 1 °C/s, but appreciable amounts are found when the microstructure is bainitic. This is due to the presence of approximately 0.6% Si and 2–3% Ni, and the very slow cooling rate involved.

3. Prior austenite grain size and the 90th percentile grain size in the cumulative grain area distribution (D90%) decreased with increasing cooling rate because of the concomitant reduction in the time for grain growth between the end of austenitization and the beginning of phase transformation.

4. Refinement of the martensite lath size and effective grain size with increasing cooling rate leads to an increase in UTS, YS, YS/UTS ratio and hardness in both investigated steels. Low toughness at the highest cooling rate (60 °C/s) is due to the very high strength combined with less retained austenite compared to slower cooling rates. At the lowest cooling rate (0.01 °C/s), low toughness despite low strength is due to a large PAGS, a large D90%(GA), the presence of prior austenite grain boundary carbides, and the appearance of granular bainite.

5. The steel with the higher C and Cr contents showed higher hardness and strength but with slightly lower toughness and elongation to fracture compared to the steel with the higher Ni content. The larger effective grain size of the steel with the higher C and Cr contents is compensated by higher solid solution and precipitation strengthening, thereby leading to higher strengths than obtained with the high-Ni steel.

6. The lower contents of C and Cr in the high-Ni steel led to lower volume fractions of proeutectoid carbides at PAG boundaries at the slowest cooling rate studied.

7. The best combination of strength and CVN impact toughness is achieved by choosing cooling rates that achieve a fully martensitic microstructure without the formation of granular bainite and proeutectoid grain boundary carbides.

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