Microstructure, Mechanical and Corrosion Properties of Extruded Milled Magnesium Powder

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Magnesium materials are interesting for application in medicine as biodegradable implants. There is an effort to improve mechanical and corrosion properties for this application. Powder metallurgy seems to be a progressive method suitable for improving those demanded properties. Therefore, this paper deals with the preparation of pure Mg by extrusion of milled powder. The milling process should lead to better homogeneity of microstructure and the disturbing of the oxide layer on the powder particles. Also, the input deformation energy in the milled powder should affect the deformation and recrystallization process during extrusion. In this paper, the influence of extrusion temperature on microstructure, mechanical, and corrosion properties is evaluated. Higher extrusion temperature leads to the larger deformed grains in the extrusion direction and higher tensile strengths. On the other hand, the plasticity and compressive yield strengths are reduced with higher extrusion temperatures. Corrosion properties are negatively affected by the iron inclusions incorporated in the structure during milling. Otherwise, corrosion resistance decreases with increasing extrusion temperature due to the grain size.

Keywords: Magnesium, extrusion, mechanical properties, corrosion, powder metallurgy.

1 Introduction

Lightweight materials such as magnesium are suitable for application in the automotive and aviation industry. There are high demands on the mechanical properties of such materials [1, 2]. Another utilization of Mg is in the field of medicine where it serves as biodegradable implants. Mg is a suitable element for such applications due to its good biocompatibility and relatively good mechanical properties together with the ability of degradation in organisms without the creation of toxic corrosion products. Similar properties of Mg with the bone tissue result in the absence of the problem known as stress shielding effect which is common for materials that have far higher Young’s modulus than bone tissue [3, 4]. The disadvantage of magnesium for such an application is its high corrosion rate which is associated with hydrogen release and the rise of pH near the implant. This might lead to the complications with healing process. Therefore, magnesium is usually alloyed for the improvement of mechanical and corrosion properties. Nevertheless, alloying elements have to be carefully chosen in order not to deteriorate biocompatibility.

From the biocompatibility point of view, it is better to use pure Mg. High purity Mg is characterised with good corrosion resistance as there is an absence of galvanic corrosion. Otherwise, pure Mg alloys with common purity are sensitive for impurities such as Fe, Ni, Cu, and Co. Dangerous impurities have to be held beneath certain limits in order to prevent excessive localised galvanic corrosion [5, 6].

Mechanical and corrosion properties of pure Mg might be improved by processing methods. Powder metallurgy is said to lead to the generally superior properties of most alloys. Also, thermomechanical processing and associated grain refinement exert improved properties [7, 8]. Powder metallurgy is a wide subject that allows a lot of possible modifications of powder even before compactization. Powder might be heat-treated, chemically treated, or milled. Milling of the powder is based on the input of plastic deformation to the material. As a result, a powder might be finer and original grains are severely deformed. Although, the heat created during milling might result in partial dynamic recrystallization for materials with low recrystallization temperature such as Zn. Deformation which is put in the material during milling has an impact on the recrystallization processes during compactization [9]. The final structure should be very fine-grained.

One of the best methods of powder compactization is extrusion [10]. During extrusion, a green compact of the material is passed through a hole with a lower diameter, and the material is, therefore, severely deformed. Extrusion of magnesium has to be done at elevated temperatures due to the low amount of slipping planes of Mg at room temperature. At elevated temperatures, new slipping planes can be activated and makes the deformation of Mg possible. The extrusion...
temperature is an important factor which has also a great effect on the final properties of the material. Magnesium alloys are also known to form a typical basal texture after deformation. Such texture has a great effect on the anisotropy of mechanical properties due to the connection between grain orientation and twinning [11]. Milling of the powder before extrusion could lead to the suppression of the texture after extrusion [12].

Therefore, this work is aimed at the preparation of pure Mg by extrusion of milled powder at different extrusion temperatures. The impact of extrusion temperature on the microstructure, mechanical, and corrosion properties is investigated.

2 Materials and methods

2.1 Sample preparation

Magnesium commercial atomized powder (30 g) with impurities measured by ICP-MS (Elan DRC-e) (90 ppm Fe, 10 ppm Cu, 20 ppm Ni) was milled in a steel planetary ball mill for 4 hours with the speed of 400 RPM. The weight ratio between the powder and the balls was 1:19. Milled powder was pressed with a pressure of 255 MPa on LabTest 5.250SP1-VM at room temperature for 5 min. Magnesium green compacts were inserted into extrusion form and they were left for 10 min to heat up on the desired extrusion temperature. Subsequently, samples were extruded at 200 °C, 300 °C, 400 °C, and 500 °C for material extruded at 200 °C, 300 °C, 400 °C, and 500 °C respectively. The average length was 3.0 ± 1.3 µm, 16 ± 7 µm, 192 ± 75 µm, and 208 ± 74 µm for extrusion at 200 °C, 300 °C, 400 °C, and 500 °C respectively. The grains were, therefore, prolonged in the direction of extrusion in all cases. The average grain size of the samples extruded at lower temperatures was smaller compared to the extruded Mg powder without milling [14, 15].

Samples extruded at 400 °C and 500 °C were characterised by large prolonged grains in the extrusion direction. This might be associated with grain growth which could occur even before extrusion. Deformation energy which was put into the powder during milling and high temperature could cause rapid grain growth during heating up of the green compact [9]. Those large grains were subsequently extruded and therefore, they are prolonged and probably contain residual stresses. Also, high extrusion temperature enabled better deformation as new slipping systems of the HCP structure were activated [16]. Nevertheless, there are also very fine recrystallized grains usually on the surface area was 100 ml·cm⁻². After 14 days, samples were removed from the immersion solution and were rinsed in distilled water and dried. The corrosion products were removed by the solution of 200 g·l⁻¹ CrO₃, 10 g·l⁻¹ AgNO₃, 20 g·l⁻¹ Ba(NO₃)₂ at room temperature. Samples were then dried and weighed. The corrosion rate was calculated from weight changes.

3 Results and discussion

3.1 Microstructure

Microstructures of extruded milled Mg powders at different temperatures are shown in Fig.1. One can see that in each material there are up to 5 µm large iron inclusions (white bright spots), which were incorporated to the Mg matrix during milling from the milling balls and milling container. Inclusions will have probably slightly deteriorating effects on the mechanical properties and it will strongly affect corrosion properties [13]. The main difference in the microstructure of samples lies in the different grain sizes. The average thickness of the grain measured by image analysis was 1.3 ± 0.4 µm, 2.0 ± 0.6 µm, 9 ± 6 µm, and 26 ± 13 µm for material extruded at 200 °C, 300 °C, 400 °C, and 500 °C respectively. While the average length was 3.0 ± 1.3 µm, 16 ± 7 µm, 192 ± 75 µm, and 208 ± 74 µm for extrusion at 200 °C, 300 °C, 400 °C, and 500 °C respectively. The grains were, therefore, prolonged in the direction of extrusion in all cases. The average grain size of the samples extruded at lower temperatures was smaller compared to the extruded Mg powder without milling [14, 15].

The microstructure was characterised by scanning electron microscopy (SEM – Tescan VEGA3) equipped with energy dispersion spectrometry (EDS, AZtec). Samples were ground on SiC grinding papers (P80-P2500) and polished on diamond paste D3, D2, and D0.7. The final polishing was done on Etrisol E. The grain size was measured by image analysis of 6 images by ImageJ software.
edges of the large grains. Materials extruded at 200 °C and 300 °C were characterised by much finer microstructure. Both temperatures are above the recrystallization temperature of Mg, therefore, the recrystallization of green compact before extrusion could also occur. However, the stimulus energy was probably not that high because of lower temperature, therefore, the structure is much finer. Also, the extrusion temperature was too low for the activation of certain slipping systems and therefore, there was only a limited amount of prolonged deformed grains [16]. There is however a slight difference in microstructure between samples extruded at 200 °C and 300 °C. The microstructure of the sample extruded at 300 °C is more homogeneous than the sample extruded at 200 °C. This might be connected with the activation of the slipping planes and better recrystallization before extrusion for the sample extruded at 300 °C. Extrusion temperature of 200 °C is below the ductile transition temperature (225-250 °C) which is necessary for the activation of <c+a> pyramidal slip, therefore, only low temperature dynamic recrystallization (LTDRX) occurs [16]. Limited amount of slipping planes and limited stimulus energy for recrystallization leads to the predominant effect of hardening mechanisms during extrusion. Extrusion temperature of 300 °C is probably close to an equilibrium where hardening and softening processes are in balance as will be discussed later.

**Fig. 1** The microstructure (SEM) of milled Mg powder extruded at 200 °C, 300 °C, 400 °C, and 500 °C.
3.2 Mechanical properties

Representative compressive (solid line), tensile (dashed line), and extrusion (dash-point line) curves of the prepared materials are summarized in Fig.2. Extrusion curves of samples extruded at 400 °C and 500 °C have a similar shape. There is only a different onset force needed for the start of the extrusion which is associated with larger grains which were probably created during heating up of the green compact and higher extrusion temperature which activates new slipping systems and enables easier deformation [16]. After reaching the force needed for the initiation of the extrusion, the force is descending. This is associated with high temperature as softening mechanisms prevail over hardening. The situation is different for materials extruded at 200 °C and 300 °C. Extrusion temperature of 300 °C is characterised with an almost constant force of extrusion, which might mean that the softening and hardening mechanisms are at equilibrium. On the other hand, extrusion temperature of 200 °C has only ascending course of the curve, which means that hardening mechanisms prevail over softening due to the low temperature.

One can see, that the tensile yield strength (TYS) and ultimate tensile yield strength (UTS) are higher with higher extrusion temperature. On the other hand, the plasticity is highly deteriorated with higher extrusion temperature. This is associated with grain size, grain orientation, and residual stresses from the milling and extrusion process. The grain sizes of samples extruded at higher temperatures are larger and according to the Hall-Petch relation, the TYS should be lower. In this case, there could be texture and residual stress in the deformed stretched grains. As a consequence of texture and residual stress, the TYS and UTS are improved at the cost of lower plasticity. In this case, the orientation of grains is unfavourable for the twinning mechanism and, therefore, the deformation is primarily provided by a slip mechanism that requires more energy than the twinning mechanism. Contrary, the twinning mechanism occurred during compressive testing [17-19]. Such behaviour could be observed on compressive curves (solid line) of the samples extruded at 400 and 500 °C. The twinning occurs after reaching the compressive yield strength (CYS) and resulting in the hardening mechanism of the newly formed twins. The twinning is also amplified by larger grains. This hardening was slightly visible for the sample extruded at 300 °C and was missing for the sample extruded at 200 °C. This is associated with finer grains and the lack of deformed prolonged grains with a specific texture that would cause the twinning. Moreover, TYS of products extruded at 200 °C and 300 °C are lower compared to the CYS, which is common for most of the materials as possible defects have a greater impact on tensile properties. Therefore, there is no observed anisotropy of mechanical properties caused by the texture.

Mechanical properties of prepared materials were superior compared to extruded pure Mg prepared by conventional casting with subsequent extrusion (TYS = 106 MPa, UTS = 178 MPa, A = 10 % - measured by Lei et al.[20]; TYS = 108 MPa, UTS = 173 MPa, A = 6.7 %, CYS = 120 MPa, UCS = 190 MPa – measured by Kubasek et al. [21]) probably due to the smaller grains size. Milling of the powder prior to extrusion provides comparable tensile properties as the powder metallurgy products (TYS = 232 MPa, UTS = 259 MPa, A = 4.3 % - measured by Peréz et al. [15]; TYS = 235 MPa, UTS = 273 MPa, A = 6.8 % - measured by Cavojsky et al. [14]). Nevertheless, those authors only measured tensile properties, and therefore, the effect of texture on the anisotropy of the mechanical properties cannot be discussed.

![Fig. 2 A) Extrusion curves (dash-point line), B) Compressive (solid line) and tensile (dash line) of prepared materials.](image-url)
3.3 Corrosion properties

Corrosion properties were measured in the SBF at 37 °C for 14 days and the final corrosion rates after removal of corrosion products are summarized in Tab.1. One can see that the corrosion rate of all products is relatively high due to the presence of Fe from ball planetary mill. Nevertheless, the values are still lower compared to extruded gas-atomized powder. This is due to the disruption of the oxides on the surface of the powder during milling. Therefore, the specific microstructure with a continuous network of oxides, which greatly deteriorate corrosion properties is not present in the sample [22, 23]. Moreover, the same trend that the corrosion rate increases with increasing extrusion temperature was observed. This is associated with a finer structure which improves corrosion resistance. Measured corrosion rates are still better than an as-cast ingot of Mg (9 mm·a⁻¹) [24].

| Tab.1 Corrosion rate in simulated body fluid |
|---------------------------------------------|
| Ex 200 °C | Ex 300 °C | Ex 400 °C | Ex 500 °C |
| v_cor [mm·a⁻¹] | 1.9 ± 0.1 | 3.4 ± 0.2 | 5.8 ± 0.2 | 6.0 ± 0.3 |

4 Conclusion

Rods of pure Mg prepared by extrusion of milled powder at 4 different temperatures were successfully created. It was investigated that higher extrusion temperature resulted in a structure with large deformed grains, while lower extrusion temperature exerted fine-grained structures. Large grains probably combined with texture caused anisotropy of mechanical properties of products extruded at 400 °C and 500 °C. Higher extrusion temperature led to the greater TYS and UTS, however, at the cost of lower elongation. Materials extruded at higher temperatures exerted higher UTS than materials extruded at 200 or 300 °C due to the possible hardening. Extrusion temperature of 300 °C seems to exert a balance between dynamic hardening and softening processes. Therefore, it is characterised with the best combination of mechanical properties out of the tested materials. The corrosion rate of all materials was negatively affected by the iron inclusions. Nevertheless, materials extruded at lower temperature exerted lower corrosion rate due to the smaller grain size.

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