Oxidation therapy: the use of a reactive oxygen species-generating enzyme system for tumour treatment

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Summary  Oxygen radicals induce cytotoxicity via a variety of mechanisms, including DNA damage, lipid peroxidation and protein oxidation. Here, we explore the use of a polyethylene glycol (PEG)-stabilised enzyme capable of producing reactive oxygen species (ROS), glucose oxidase (GO), for the purpose of harnessing the cytotoxic potential of ROS for treating solid tumours. PEG-GO (200 U), administered by two intratumoral injections 3 h apart, produced a significant growth delay in subcutaneous rat 9L gliomas as compared with control animals receiving heat-denatured PEG-GO. Rats were protected from systemic toxicity by subsequent i.v. administration of PEG-superoxide dismutase (PEG-SOD) and PEG-catalase. In vivo tumour metabolic changes, monitored using 31P magnetic resonance spectroscopy (31P-MRS) 6 h following initial administration of PEG-GO, revealed a 96 ± 2% reduction in the ATP/P ratio and a 0.72 ± 0.10 unit decline in intracellular pH. A 3-fold sensitisation of 9L glioma cells in vitro to hydrogen peroxide could be achieved by a 24 h preincubation with buthionine sulfoximine (BSO). This study suggests that oxidation therapy, the use of an intratumoral ROS-generating enzyme system for the treatment of solid tumours, is a promising area which warrants further exploration.

Hydrogen peroxide is produced in mammalian cells during normal metabolism by different oxidases and from either spontaneous or catalysed superoxide dismutase (SOD)-catalysed superoxide anion dismutation. Hydrogen peroxide can readily cross cellular membranes, and is the most sensitive cellular target structure for hydrogen peroxide is DNA, in which single-strand breaks occur (Hoffman & Meneghini, 1979; Bradley & Erickson, 1981; Imlay et al., 1988). This genotoxicity has been proposed to be the result of transition metal-driven Haber–Weiss reactions leading to the generation of cytotoxic hydroxyl radicals (Filho & Meneghini, 1984). Over the past 40 years there have been many attempts to treat tumour-bearing patients or animals with preformed hydrogen peroxide (Hollcroft et al., 1952; Makino & Tanaka, 1953; Turner, 1953; Green & Westrop, 1958; Sugiuira, 1958; Kremetz et al., 1963; Mealey, 1965; Kaibara et al., 1971). In these reports, hydrogen peroxide was injected directly into the solid tumour, the circulation or the peritoneal cavity of mice with malignant ascites. The results from these studies, however, were anecdotal. It was not until the early 1980s that investigators reported that hydrogen peroxide can exert a direct anti-tumour effect in vivo and thus prolong host survival (Nathan & Cohn, 1981). In that previous study, glucose oxidase (EC 1.1.3.4) was conjugated to carboxylated latex microspheres and was found to have an anti-tumour effect both in the peritoneal cavity and when injected into the tumour bed of subcutaneously implanted P388 cells at the time of cell implantation (Nathan & Cohn, 1981). The microspheres were at the tumour site and hydrogen peroxide generation was achieved through the following reaction:

GO

β-D-Glucose + O2--------- > d-glucono-1,5-lactone + H2O2

It is known that enzymes can be stabilised in vivo by attaching PEG units to the lysine residues. Enzyme modification in this manner increases in vivo half-life from 6 min to 30–40 h, reduces the antigenicity of the native protein and inhibits proteolysis (Pyatak et al., 1980). To our knowledge, there have been no reports demonstrating growth inhibition of a solid tumour using hydrogen peroxide or a hydrogen peroxide-generating system. The unrealised potential of this treatment was the motivating factor for this present study, in which we describe the anti-tumour effect of PEG-GO on the growth of cultured 9L glioma cells and solid s.c. 9L tumours in rats. In addition, results using 31P-MRS to examine the effects of this therapeutic approach on tumour phosphate metabolites and pH in vivo are presented.

Materials and methods

Preparation of PEG–glucose oxidase (PEG-GO)

The preparation of PEG-catalase, PEG-SOD and PEG-GO was accomplished using previously reported procedures for PEG-SOD (Pyatak et al., 1980). In brief, PEG-GO was prepared by addition of 40 mg of protein (Aspergillus niger) in 10 ml of 0.1 M boric acid, pH 9.8. Activated PEG (1.0 g; M; 5,000, Sigma, St Louis, MO, USA) was added and the mixture was stirred for 1 h. Unattached PEG was removed by five sequential dialysis steps against equivalent volumes of phosphate-buffered saline (PBS) (calcium chloride, 0.9 mm; potassium chloride 2.7 mm; potassium dihydrogen phosphate 1.5 mm; magnesium chloride 0.3 mm; sodium chloride 137.0 mm; disodium hydrogen phosphate, 7.5 mm) using an Amicon (Beverly, MA, USA) ultrafiltration chamber (model 8050) equipped with an XM-50 membrane. The Mf activity of PEG-GO were determined by laser desorption linear time-of-flight mass spectrometry (Vestec Model VT 2000) and spectrophotometry (Michiels & Remacle, 1988) respectively. PEG-SOD and PEG-catalase were assayed for activity using standard spectrophotometric methods (Pyatak et al., 1980; Michiels & Remacle, 1988).

Sulforhodamine B (SRB) cell toxicity assay

Rat 9L glioma cells were grown until confluent as monolayers in 75 cm² sterile plastic flasks in modified Eagles minimum essential medium containing 10% fetal calf serum at 37°C in a humidified atmosphere containing 95% air and 5% carbon dioxide. Cells were harvested by trypsinisation, counted, diluted in serum-free medium, plated in 96-well culture plates at a density of 1,000 cells per well and allowed to grow for 24 h before treatment. Increasing concentrations of PEG-GO and hydrogen peroxide were subsequently added to the 9L cells in Krebs–Ringer bicarbonate buffer (KRB) sodium chloride 113.0 mm; sodium bicarbonate 29.0 mm; potassium chloride 4.6 mm; magnesium sulphate 0.5 mm; disodium hydrogen phosphate 0.6 mm; potassium...
hydrogen phosphate 0.3 mm; potassium bicarbonate 1.0 mm; calcium chloride 2.5 mm; glucose, 5.5 mm; pH = 7.4) for a 60 and 30 min exposure period respectively. The activity of PEG-GO was determined in KRB on the same day of cell exposures to ensure accurate dosing. An automated microculture assay using SRB (a protein-binding dye; Rubinstein et al., 1990) was used to quantitate the toxicity of preformed and PEG-GO-generated hydrogen peroxide on cultured 9L glioma cells. Under the conditions used in this study, the optical absorbance was directly proportional to the number of 9L cells. Following hydrogen peroxide exposure, cells were gently rinsed three times and allowed to grow for 4 days. The cells were then fixed and stained, and the absorbance values were obtained using an AutoReader microtitre plate reader (Cayman Chemicals, Ann Arbor, MI, USA). Toxicity was assessed in terms of fractional cell survival relative to control.

Studies to evaluate the effects of reduced intracellular glutathione (GSH) levels on sensitivity of 9L tumour cells were done by preincubating cells for 24 h in the presence of 0.1, 0.5, 1, 2 and 5 mM buthionine sulfoximine (BSO) before exposure to increasing concentrations of hydrogen peroxide as described above. GSH levels were determined spectrophotometrically using a kit provided by Bioxytech (Marne, France).

In vivo evaluation of PEG-GO
Specific pathogen-free male Fischer-344 rats were purchased from Charles River Breeding Laboratories (Wilmington, MA, USA). Rats weighed 275–300 g at the beginning of the experiment and were housed in the university laboratory animal medicine facility. All procedures were approved by the Institutional Animal Care and Use Committee. Subcutaneous gliomas were induced by injection of 1 × 10^6 9L glioma cells in 0.2 ml of serum-free culture medium over the right thigh muscle under halothane anaesthesia. Caliper measurements of tumour dimensions were repeated every 2 days beginning on the eighth day post cell implantation. Tumour volumes were determined using the equation: Tumour volume = 0.5 (length × width^2). Rats with subcutaneous gliomas were treated 12 days after cell inoculation, at which time the mean tumour volume was approximately 0.7 cm^3. PEG-GO (50 or 200 U) contained in 50 μl of phosphate-buffered saline (PBS) was administered by an intratumoral injection. Tumours treated with 200 U of PEG-GO received a second identical injection 3 h later. Care was taken to distribute the PEG-GO as evenly as possible throughout the tumour mass. At the conclusion of the second 200 U PEG-GO injection, 10,000 U each of PEG-SOD and PEG-catalase were administered i.v. in order to protect the animals from systemic toxicity owing to the potential leakage of hydrogen peroxide and/or the PEG-GO from the tumour site into the vascular system. Control animals were treated identically except that the PEG-GO was denatured by heating at 90°C for 15 min prior to injection.

In order to determine the effects of this therapy on cellular energy state in vivo, 31P-MR spectra were recorded on several animals before and 6 and 24 h after intratumoral injection using a horizontal-bore 7 T Spectroscopy Imaging System magnetic resonance instrument. For MRS experiments, rats were anaesthetised by an i.p. injection of a mixture containing ketamine (80–100 mg kg^-1) and xylazine (13 mg kg^-1). 31P-MR spectra were recorded using a single-turn 9-mm diameter surface coil probe at 121.4 MHz using the following acquisition parameters: sweep width = ± 3,000 Hz; number of scans = 1,000; interpulse delay = 2.2 s; number of data points = 4096. 31P-MR spectra were apodised with a 30 Hz exponential line-broadening function prior to Fourier transformation. Quantification of ATP and inorganic phosphate (Pi) resonance areas was accomplished using a spectral deconvolution routine and pH was calculated from the chemical shift of P, relative to phosphocreatine (PCr). When the PCr resonance was absent, the water resonance from the proton spectra obtained during shimming procedures was used to calculate the position of PCr and hence pH as previously described (Madden et al., 1991). Determination of pH values using this approach has been reported to be well within the expected accuracy of the NMR pH measurement of ± 0.05–0.1 pH units (Roberts et al., 1981).

Statistical analysis
Piecwise linear regression (Wasserman & Kutner, 1990) was used to model growth in tumour volume as a function of time, allowing for a different rate of growth in the treated group following treatment on day 12. The natural logarithm of tumour volume was used as the dependent variable because preliminary analysis indicated that the tumour volume increases in an approximately exponential manner with increasing time. The independent variables in the model were time in days since implantation and, for the treated group, time since treatment began. This variable was equal to zero both for the control group and for the treated group before treatment.

For 31P-MR data, all values are reported as means ± s.e.

Results
Characterisation of PEG-GO
The average Mf of native GO and PEG-GO was determined to be 150 and 185 kDa respectively. Because the PEG used in this study had a Mf of 5,000, an average of seven PEG moieties were attached to each PEG-GO molecule. Attachment of PEG reduced the enzymatic activity by 40% compared with native glucose oxidase as estimated from Lineweaver–Burk plots (data not shown).

In vitro cytotoxicity studies
Survival studies of 9L glioma cells 4 days following exposure to increasing activities of PEG-GO for 60 min revealed that this enzyme system is cytotoxic to 9L glioma cells (Figure 1a). Cytotoxicity studies of 9L glioma cells following a 30 min exposure to preformed hydrogen peroxide revealed a dose-dependent toxicity with an IC50 = 60 μM (n = 16, Figure 1b). In the presence of 0.1 mM BSO, an inhibitor of GSH synthesis, the IC50 decreased to 22 μM hydrogen peroxide, indicating an approximate 3-fold increased sensitivity to hydrogen peroxide (Figure 1b). Cell toxicity following a 24 h incubation with 0.5, 1, 2 and 5 mM BSO was identical to that observed for 0.1 mM BSO. Spectrophotometric determination of intracellular GSH levels under these conditions revealed a decrease from 42 to 20 nmol mg^-1 protein with 0.1 mM BSO. At BSO concentrations of 0.5 mM or higher, GSH was not detectable.

In vivo 31P-MRS and growth inhibitor studies
Shown in Figure 2 are in vivo 31P-MR spectra of a subcutaneous 9L glioma acquired before PEG-GO injection (Figure 2a) and 6 h (Figure 2b) and 24 h (Figure 2c) following intratumoral injection of 50 U of PEG-GO. The spectrum obtained 6 h post intratumoral injection of 50 U PEG-GO shows a significant increase in the energy state as indicated by the decrease in the PCr and ATP resonance areas and the increase in P. However, after 24 h, the energy state almost completely recovered and the rate of tumour growth remained unaltered. Shown in Figure 3 are representative in vivo 31P-MR spectra of subcutaneous 9L gliomas acquired before tumour treatment (Figure 3a) and 6 h following intratumoral injection of 1 × 200 U of denatured PEG-GO (Figure 3b) and active (2 × 200 U) PEG-GO (Figure 3c). Intratumoral injection of heat-denatured PEG-GO (Figure 3b) resulted in a 34 ± 17% (n = 3) decline in tumour ATP/P levels and a reduction in tumour pH by 0.10 ± 0.10 U (n = 3). Administration of active PEG-GO resulted in the nearly complete loss of PCr and ATP and a reciprocal increase in P (Figure 3c). For PEG-GO-treated animals, the
The antioxidative enzymes disappeared cutaneous limb, lethargy and ultimately...

**Discussion**

The significant delay in tumour growth demonstrates the feasibility of oxidation therapy, which we define as the use of an intratumoral ROS-generating enzyme system for the treatment of solid tumours. Previous studies using direct injections of hydrogen peroxide into solid tumours were unsuccessful (Green & Westrop, 1958). A study using glucose oxidase conjugated to latex microspheres did not report attempts at treating solid tumours but rather tumour cells in the peritoneal cavity and co-injection of tumour cells with the microspheres at the time of s.c. implantation (Nathan & Cohn, 1981). However, modification of GO and PEG resulted in only a 40% reduction in activity as compared with a 90% loss of enzyme activity when conjugated to latex microspheres (Nathan & Cohn, 1981). This reduction in activity, when combined with the increased volume taken up by the inert microsphere material, resulted in a net 4,000-fold decrease in total enzyme activity per volume as compared with the PEG-GO system. Although the latex microsphere system has the advantage of retaining the GO at the tumour site, the therapeutic effectiveness towards solid tumours (such as the 9L glioma) is probably limited since the injectable volume necessary to attain the dose used in this present study would exceed the tumour volume.
In this study, \(^{31}\text{P}-\text{MRS}\) was used to assess the \textit{in vivo} metabolic effect of oxidation therapy since two distinct mechanisms by which hydrogen peroxide can deplete cellular ATP levels are known. The first mechanism involves activation of poly(ADP-ribose) polymerase following hydrogen peroxide-induced DNA strand breaks (Schaftafet et al., 1986). Activation of poly(ADP-ribose)polymerase consumes NAD\(^+\), a process that is subsequently associated with a loss of ATP levels (Ueda & Hayaishi, 1985; Berger et al., 1986; Bruchet et al., 1991). This second mechanism involves hydrogen peroxide inhibition of glycolysis by the inactivation of glyceraldehyde 3-phosphate dehydrogenase resulting in depletion of ATP levels (Brodie & Reed, 1987). Our results, obtained from \textit{in vivo} \(^{31}\text{P}-\text{MRS}\) evaluation of treated gliomas, are consistent with the two mechanisms mentioned above in which ATP depletion is directly linked to oxidative stress. \(^{31}\text{P}-\text{MR}\) spectra of PEG–GO-treated 9L gliomas revealed a dramatic decline in both the ATP/P\(_i\) ratio and tumour pH as compared with the control group at 6 h post treatment. The exact mechanism(s) for these changes could be one or a combination of the two aforementioned, however other possibilities, such as a decrease in tumour blood flow, cannot be ruled out. Whatever the exact mechanism(s) for cell death, it is important to note that metabolic changes preceded the inhibition of tumour growth, suggesting that \(^{31}\text{P}-\text{MRS}\) can provide a non-invasive means for following the progress of the treatment by monitoring ATP levels. This capability may be very useful during the actual treatment period by providing the feedback necessary to determine when the generation of ROS could be discontinued.

Optimisation of this therapeutic approach will require improvements in the delivery and control of the ROS-generating enzyme system and assessment of the pathological consequences. Glucose oxidase was chosen to demonstrate the concept of oxidation therapy because it has a relatively high specific activity, is commercially available and because the substrates necessary for hydrogen peroxide production, glucose and oxygen are abundant in tissue. However, the enzyme has a fundamental disadvantage in that its activity cannot be directly controlled, necessitating administration of large doses of antioxidative enzymes (or antioxidants) in order to protect the rat from systemic toxicity. This tissue may be exacerbated in the rat because blood catalase activity in this species is only 25% of that in the human (Lorincz et al., 1943). This disparity between species is also reflected in the fact that human blood foams in contact with hydrogen peroxide, whereas this phenomenon does not occur when hydrogen peroxide is added to rat blood. Oxygenation of rat blood does not occur, but rather the blood turns dark-brown owing to the formation of methaemoglobin. Methaemoglobin formation was observed in the present studies in treated rats that were not protected by administration of PEG–catalase and PEG–SOD.

An alternative oxidase enzyme system in which the activity of the hydrogen peroxide generating system could be manipulated by substrate availability and/or direct chemical inhibition of the enzyme would provide a significant advantage over glucose oxidase. This would reduce excessive oxidative stress on surrounding normal tissues and blood. This capability would require that the substrate is not endogenous to mammalian tissue or present in low enough concentrations as not to induce significant formation of hydrogen peroxide. A specific example of an enzyme with these properties is \(\text{D-amino acid oxidase (DAAO)}\) (EC 1.4.3.3). This enzyme catalyses the following reaction and can be manipulated by substrate depletion or chemical inhibition:

\[
\text{DAAO} \quad \text{D-Amino acid} + \text{H}_2\text{O} + \text{O}_2 \rightarrow >\text{2-oxo acid} + \text{NH}_3 + \text{H}_2\text{O},
\]

Therefore, the long-term development and optimisation of oxidation therapy will necessitate the use of an oxidase in which the activity can be manipulated.

Selectivity of oxidation therapy for gliomas will depend on the relative sensitivity of neurons, astrocytes and glialoma cells to hydrogen peroxide-induced toxicity. We have obtained...
preliminary in vitro results showing the IC50 of a 30 min exposure to hydrogen peroxide for primary rat neuronal and primary astrocytic cultures to be 98 and 585 μM respectively (O. Ben-Yoseph, P. Boxer & B.D. Ross, unpublished results), as compared with 60 μM for cultured 9L cells. We have also shown in this present study that sensitivity to hydrogen peroxide could be greatly increased (3-fold) when cultured 9L tumour cells were pretreated with BSO, (Griffith, 1982). BSO has an additional advantage of selectivity as it has been shown to reduce the GSH levels of intracranial gliomas without affecting normal brain GSH levels (Lippitz et al., 1990). Thus, pretreatment of the host with BSO should provide an excellent method for the selective sensitisation of malignant brain tumours to radiation therapy. Further methods for sensitising gliomas to oxidative stress include, for example, inhibition of tumour antioxidant defences or DNA repair mechanisms.

Future development of oxidation therapy will entail not only optimising the oxidase system and selective presensitisation of the tumour, but also improving methods of delivery to the tumour site. For treatment of malignant brain tumours, the disrupted blood–brain barrier should also allow an i.v.-administered PEG–oxidase system selective access to the glioma tissue. Intrathecal administration of the PEG–oxidase substrate could provide for activation of the enzyme predominantly within the tumour tissue. Progress of the therapy could be monitored using localised 31P-MRS in vivo.

There are a multitude of additional variations on how oxidation therapy could be utilised for tumour treatment. Possible delivery methods include the use of liposomes, biodegradable polymers or attachment to monoclonal antibodies (Philpott et al., 1980). Although the type of delivery vehicle will no doubt vary depending upon the tumour type and host organ targeted, our initial promising data provide evidence that oxidation therapy is a strong candidate for tumour treatment and warrants further evaluation.

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