Review

Anti-inflammatory potential of edible ornamental flowers grown in containers due to high anthocyanin content

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Abstract: Flowers have always accompanied people thanks to their manifold aesthetic properties. Some species have also become a component of human diet. Recent years have seen an increased interest in edible flowers and, consequently, research has been undertaken to determine their chemical composition. Dyes abundantly contained in flowers, whose role is to attract pollinating animals, are recognized substances with health promoting properties. Anthocyanins are a group of dyes that is very common in petals and other parts of flowers. Studies carried out in the twentieth and twenty-first century have found very strong antioxidant and anti-inflammatory properties of anthocyanins. Therefore, flowers used by humans for centuries to decorate their surroundings may become an easily available source of nutrients and health-promoting substances. This paper discusses the health-promoting properties of anthocyanins and collects literature on anthocyanin content in edible flowers commonly grown on balconies, terraces and roofs.

Keywords: agroecosystem, urban ecology, perennial and annual flowers, anti-inflammatory, antioxidant properties, biological activity

1. Introduction

Increased lifespan and a better quality of life have dramatically improved life expectancy of the world population. This tendency is especially visible in Western countries, but at the same time, the high availability of hypercaloric food and the increase in consumption of highly processed foods have led to the massive occurrence of chronic non-communicable diseases – mainly cardiovascular, metabolic and neurodegenerative diseases – in those countries [1]. In 2016, the World Health Organization (WHO) estimated that approximately 650 million adults were obese [2]. For these reasons, particular emphasis should be placed on increasing the consumption of fresh food containing bioactive compounds, as these substances provide health protection when interacting at many levels. Among fresh foods, especially plant-based foods contain a lot of bioactive compounds, such as polyphenolic compounds, which modulate processes occurring in the human body and have antioxidant, anti-inflammatory, anticancer, and neuroprotective effects, and can modulate glucose levels [3].

The growing demand for new nutraceutical plant food has sparked interest in edible flowers. Various flower pigments, formed in the process of evolution to attract pollinator organisms, have been shown to have high antioxidant activity, which can be a remedy for diseases of civilization [4]. Anthocyanins play an important role in the attraction strategy involving the use of colour, but their strong antioxidant potential also makes flowers an important resource, the use of which should be increased in cultivation and nutrition [5]. Flowers commonly grown by humans and thus often present in the human environment,(for example, planted every year in containers on balconies, terraces and roofs), due
to the high content of biologically active substances, could become something of a “home pharmacy” helping to fight modern diseases.

2. Chemistry and Biochemistry of Anthocyanins

The word ‘anthocyanin’ derives from two Greek words: anthos, which means flowers, and kyanos, which means dark blue [6].

Anthocyanins are secondary metabolites in land plants that contribute to the colour of leaves and flowers [7]. These pigments are primary blue, red and purple. They are synthesized via the flavonoid pathway, which is part of the general phenylpropanoid pathway [8]. The entry to the biosynthesis of phenylpropanoids is the shikimate pathway. In this pathway plants biosynthesise in three steps hydroxycinnamic acids and their derivatives, which are the precursors for a large variety of aromatic metabolites [9]. The next step to the synthesis of anthocyjanidins is the conversion of a chorismic acid to phenylalanine by the enzyme phenylalanine ammonia-lyase [10] from which cinamic acid is formed. The conversion of cinamic acid to anthocyanins requires a series of reactions: the first reaction is catalyzed by cinnamate 4-hydroxylase to form a coumaric acid and by 4-hydroxy-cynnamoyl CoA ligase to create 4-Coumaroyl CoA, which is a direct precursor to kaempferol. After four steps of enzymatic reaction from 4-Coumaroyl CoA the leucoanthocyanidins are formed [11]. By the catalysis of anthocyanin synthase (ANS), the colorless leucoanthocyanidins (flavan-3,4-diols) are oxidized to the coloured anthocyanidins [9]. Flavan-3,4-diols, also known as leucoanthocyanidins, are not particularly prevalent in the plant kingdom, instead being themselves precursors of flavan-3-ols (catechins), anthocyanins, and condensed tannins (proanthocyanidins) Anthocyjanidins are unstable under physiological conditions so they are immediately glycosylated in the 3-OH positions by UDP-glucose-flavonoid 3-O-glucosyltransferase (UFGT) to form the more hydrophilic and stable anthocyanins [12].

It has been experimentally demonstrated that all anthocyanin pigments are derived from one of three aglycones (pelargonidin, cyanidin and delphinidin). The differences in the colour of anthocyanins result from the pattern of hydroxylation and methylation, and the amount and type of sugars [13]. Anthocyanins display different colours (red, blue and purple) depending on their accumulation and chlorophyll complementary light absorbance. At low pH values, anthocyanins are present as flavillyum cations (oxonium charged oxygen), while under neutral conditions uncharged quinones are formed [14]. At pH around 2.0-3.5 anthocyanins have a pink-coral colour, while at 5.5-6.5 they are blue to purple [13]. The chromophore of conjugated double bonds carrying a positive charge on the heterocyclic oxygen ring is responsible for the intense red-orange to blueviolet colour produced by anthocyanins under acidic conditions [15].

There were 635 identified anthocyanins in 2010 [16]. Anthocyanins are present in nature mainly in the form of heterosides. The aglycon form of anthocyanins are called anthocyanidin. The basic structure of anthocyanins is composed of flavylum cation (C6-C3-C6), which could be linked to different sugars or hydroxyl or methoxyl groups [17]. The most abundant anthocyanins are delphinidin, cyanidin, petunidin, peonidin, malvidin, and pelargonidin. Glucose is the most common sugar attached to anthocyanins, but also rhamnose, xylose, galactose, arabinose, and rutinose have been reported to be linked to these compounds [6]. Depending on the number of attached sugars, anthocyanins can be mono-, di-, or tri-glycosides [17]. The presence of sugars gives more stability and water solubility than their corresponding glycosides [9]. Glycosylation primarily at the C-3 residue results in reduced maximum wavelength absorption [18].

Sugar residues may be further acylated with cinnamic acids, such as p-coumaric, ferulic, and sinapic acid, as well as aliphatic acids, such as acetic, malonic, and oxalic acid [19].
3. Antioxidant and anti-inflammatory activity of anthocyanins

3.1. Antioxidant activity

The antioxidant potential of anthocyanins depends on the ring orientation (which determines the ease with which a hydrogen atom from a hydroxyl group can be donated to a free radical), the ability of the anthocyanin to support an unpaired electron [20], the number of free hydroxyls around the pyrone ring and their positions, and the presence of other types of radicals in the main structure [21]. The protection of these pigments against oxidation processes depends on their structures. Principally, the antioxidant activity of anthocyanins is associated with the number of free hydroxyls around the pyrone ring. Higher antioxidant activity is due to the number of hydroxyls [20].

Individual anthocyanins differ in their ability to remove highly active radicals depending on the radical. For instance, pelargonidin is the most efficient against the hydroxyl radical, whereas delphinidin is the most active against the superoxide anion [22]. Free radical damage contributes to the aetiology of many chronic diseases and thus antioxidants may have beneficial effects on human health at different levels [23]. Improving the diet through the consumption of products containing natural antioxidants is one of the best strategies to create a balance between the activity of free radicals and the antioxidant system in human body [24].

The antioxidant capacity of consumed products can be measured using chemical, in vitro methods generally performed on extracts. The literature mentions nearly 20 methods [9, 11, 13], but in general, if there are many methods, none of them is perfect. In addition, we must remember that the indicators give us a picture of the potential of the product, but they will not answer the question of how many substances will be absorbed and what impact they will have on the body. The most popular methods of measuring antioxidant potential are based on the ability to bind free radicals (DDPH, ABTS), to reduce cupric or ferric ions (FRAP, CUPRAC), to protect a target molecule exposed to a free radical source (ORAC, TRAP) and to inhibit the oxidation of low density lipoprotein (LDL) [25]. Antioxidant capacity is a function of the content and types of phytochemicals that are present in fresh tissues. However, individual groups of compounds may differ considerably in terms of antioxidant properties. Many studies indicate that phenols and flavonoids contribute more strongly to antioxidant capacity than ascorbic acid, vitamins, carotenoids, and other compounds [26]. Anthocyanin molecules, due to their structure, stand out from flavonoids as a group of compounds exhibiting very high antioxidant capacity [27]. Still, some research suggests that the bioavailability of anthocyanins is lower than that of other flavonoids. Anthocyanins were initially perceived as poorly absorbed and metabolized compounds, which cast doubt on whether they could have a biological effect in humans. They were found only in the plasma in their intact form (glycosylated). However, most of those studies were based on plasma and urine analysis for anthocyanin metabolites derived from glucuronidation and sulphation metabolism. More recent studies increasingly allow to identify metabolites of anthocyanins at higher concentrations than the parent compounds [28]. According to some research, anthocyanins may be metabolized by intestinal microflora producing a group of new products that have not yet been identified, not to mention quantified. In addition, recent studies indicate that anthocyanins are rapidly absorbed, with maximum plasma concentration (Cmax) of between 45 min to 4 h after ingestion of a meal containing anthocyanins, depending on the conditions of the trial. When anthocyanins were ingested alone and after a night, Cmax was reached after only 1 h [29], but if they were consumed together with other food, the absorption decreased; especially if food contained fat, Cmax was reached only after 4 hours [30]. The structure of anthocyanins affects their absorption by the human body. It has been shown that 3-monoglucosides of anthocyanidins are less bioa-
vailable than their corresponding rutinosides [31]. Also, the absorption differences between malvidin and petunidin may be due to the fact that a large number of hydroxyl groups in the molecule decreases its bioavailability. However, it should be remembered that the absorption capacity will also depend on the amount of anthocyanins and the presence of other compounds. It was found out that the ingestion of anthocyanins together with sugar slowed down their absorption while the consumption of anthocyanins together with alcohol significantly accelerated their intake [32].

3.2. Anti-inflammatory activity and protection against chronic diseases

The health benefits of anthocyanins have been studied in a variety of models, ranging from human clinical trials to animal and cell culture screening to epidemiological studies [17]. The human body is in constant contact with external factors that can cause various types of damage, irritation or allergies [9] often leading to inflammation. Inflammation is a complex set of relationships between soluble compounds that can arise in any tissue in defensive response to traumatic, infectious, post-ischemic, toxic, or autoimmune injury. It is typically induced by microbial infections, but can also be triggered by tissue injury or trauma that occurs without the intervention of pathogens (sterile inflammation). The inflammation process usually leads to recovery from infection and healing [33].

Adaptive innate immune response induces rapid activity following infection. A wide range of molecular patterns are detected, commonly found in pathogens but foreign to mammals. They are called pathogen-related molecules patterns (PAMP) [34]. Such particles are lipopolysaccharides, surface phosphatidylserine, and aldehyde derivatized proteins as well modified forms of the classic risk factor for atherosclerosis, oxidatively modified low-density lipoprotein (LDL) or glycation [35]. The cellular response may be lysosomal endocytosis, degradation bound ligands. Involvement in the process of Toll-like receptors causes the activation of the nuclear factor kappa B (NF-κB), and protein kinase. It can induce increased phagocytosis, production of reactive oxygen and release of cytokines, autacoids and lipids coordinating and strengthening local inflammation answer [36], [37]. Recent research demonstrate that metabolites of anthocyanins can reduce the activation of NF-κB [38]. Protein kinases, cellular stress kinases, extracellular signal-regulated kinases, mitogen-activated protein kinases (AMPK) are another molecular target of anthocyanins and have been shown to be sensitive to anthocyanin treatment reducing downstream cellular signaling networks associated with serious diseases, such as chronic inflammation [17]. AMPK-activated protein kinase involved in cellular energy (glucose) metabolism caused diabetes appears to be one of the main targets of anthocyanins [39]. AMPK is an important regulator of energy homeostasis and is a molecular target of drugs used for the treatment of obesity and other metabolic diseases [40]. Another target of anthocyanins are thrombin receptor activating peptide and vascular endothelial growth factor, which are responsible for angiogenesis, cancer, and atherosclerotic risk [41].

The biological activity of isolated anthocyanins and anthocyanidins, or foods rich in anthocyanins, can be manifested in: prevention of cardiovascular disease [42], influence on cholesterol distribution, protection of endothelial cells from CD40-induced pro-inflammatory signaling [43], anticancer, antitumor and antimutagenic activity [44], beneficial effects in diabetes [45], protective effect against oxidative liver damage [46], protective effect on gastric inflammation and damage [47], antimicrobial and antiviral activity [48, 49], slowing down neuronal and behavioural aging [50] and protection from some neurodegenerative diseases such as Alzheimer’s disease [51]. Anthocyanins and anthocyanidins also effectively induce insulin secretion when tested in pancreatic cell lines.
The effectiveness of insulin secretion depends on the number of hydroxyl groups in the B-ring of their structures [45].

Cyanidin (C15H11O6) and its derivatives are the most common anthocyanins in flowers (Table 1). The study carried out by Samarpita and Rasool [52] suggests that cyanidin is a potent inhibitor of Interleukin (IL)-17A signaling associated pathogenesis of rheumatoid arthritis, the most common autoimmune arthropathy. Cyanidin not only effectively blocks interleukin 17A/p38 but also suppresses osteoclastogenesis. This study suggests that cyanidin has great potential as a small molecule drug to be used in clinics to treat rheumatoid arthritis patients [52]. Moreover, there is evidence that cyanidin as well as delphinidin have the chemo preventive effect against skin cancer [53].

The effect of anthocyanins on microbial pathogens has not been studied in depth up to now. However, the results obtained so far are very promising.

4. Factors influencing anthocyanin content in ornamental plants

Anthocyanin accumulation is strongly regulated by plant development and genotype, and by environmental factors [54]. One of the goals of ornamental plant breeding is to broaden the colour palette of different species by adding missing colours. For example, the best-selling cut flowers so far, namely, rose, chrysanthemum, carnation and lily, include no blue cultivars in their palette, while petunias are not red/orange [55]. Purple flowers in rose and carnation were obtained by changing the decoration pattern on the basic skeleton of anthocyanins, i.e. increasing the accumulation of delphinidin [56].

Temperature has a big impact on anthocyanin accumulations. Strong temperature variations between day time and night time favour the accumulation of soluble solids, and more soluble solids enhance the accumulation of anthocyanin [57]. However, too low temperatures slow down physiological processes and can thus also limit anthocyanin production. The plant hormone abscisic acid (ABA) has been suggested to play an important role in anthocyanin accumulation. ABA treatment increases anthocyanin content in grape skin and induces the expression of anthocyanin-biosynthesis genes [58]. Studies on the effect of altitude on anthocyanin accumulation in blueberry fruit found out that plants growing at a lower altitude accumulated more anthocyanins [54].

Fertilization also affects anthocyanin accumulations. Pre-harvest calcium treatment was shown to upregulate the expression of anthocyanin structural genes and to increase the total phenolic and anthocyanin content [59]. The accumulation of anthocyanins in the plant is also promoted by better availability of phosphorus in soil [60] and by the application of melatonin (N-acetyl-5-methoxytryptamine). It is explained that melatonin is involved in secondary metabolism, where it induces anthocyanin and flavonoid biosynthesis [61]. The plant’s growing location also plays a role because it has been shown that ultraviolet B emitted by the sun (wave length: 280–315 nm) promotes anthocyanin synthesis. This is the part of the radiation which is only partially absorbed by the ozone layer and therefore exposure to direct sunlight stimulates the formation of anthocyanins [62].

The production of anthocyanins in ornamental plant species is also enhanced by a change in the activity of flavonoid enzymes by gene modification. This method to increase the content of anthocyanin pigments was applied to petunias and torenias [63, 64]. At the genetic level, gibberellins, which are regulators of growth and development, can also interact. It was found that during the development of petunia flowers gibberellin induced the expression of some genes such as those of chalcone synthase, chalcone isomerase, anthocyanidin synthase, and dihydroflavonol 4-reductase, which are responsible collectively for corolla pigmentation [65, 66].
5. Anthocyanin content of domestic grown edible flowers

Table 1. Anthocyanin content in edible flowers grown in containers

| Flower species                  | Cyanidin | Delphinidin | Pelargonidin | Malvidin | Peonidin | Petunidin | Total               | ORAC/FRAP (TE/100g) | Source |
|---------------------------------|----------|-------------|--------------|----------|----------|-----------|---------------------|----------------------|--------|
| Ageratum houstonianum           | 27.85    | 2.99        |              |          |          |           |                     |                      | [5]    |
| Argyranthemum houstonianum      | 2.99     | 27.85       |              |          |          |           |                     |                      | [5]    |
| Begonia sp.                     | p^D      | 759.1       |              |          |          |           | 21.18*              |                      | [5], [67], [68]    |
| Bellis perennis                 | p^D      | p^D         |              |          |          |           | no data             |                      | [69]   |
| Calendula officinalis           | 22.1^7   | 3.68        |              | 58.05    |          |           |                     |                      | [5], [72], [73]    |
| Dahlia sp.                      | 121.2    | 2.65        |              | 17.6     | 257.5    | 7.8       | 17 – 24^5^9         |                      | [74], [75]         |
| Dianthus                        | 52.4     | p^D         | p^D          | 0.73     | 13.35    | 8.9       | 5.4 - 10.2^A,9      |                      | [5], [76], [77]    |
| Dendranthema                    | p^D      | p^D         | p^D          |          |          |           | 168 – 182^B         |                      | [69], [78], [79]   |
| Phaseolus cocineus              | no data  |              |              |          |          |           |                     |                      |        |
| Fuchsia sp.                     | p^D      | p^D         |              | 7.58     | 47.52    |           |                     |                      | [5], [80]         |
| Glechoma hederacea              | p^D      | p^D         |              |          |          |           |                     |                      | [81]   |
| Heliotropium oxalis             | no data  |              |              |          |          |           |                     |                      |        |
| Helichrysum                     | 419.8    |              |              |          |          |           |                     |                      | [82]   |
| Hemerocallis                    | 21.0 – 29.0^A |          |              |          |          |           |                     |                      | [83]   |
| Hibiscus sp.                    | 2080     | 5650        |              | 155 – 206^s | 83.1^5   |           |                     |                      | [84], [85], [86], [87] |
| Impatiens                       | p^D      | p^D         |              |          |          |           |                     |                      | [88]   |
| Lavandula                       | 277.60   |              |              |          |          |           |                     |                      | [81]   |
| Lobelia                         | no data  |              |              |          |          |           |                     |                      |        |
| Lobularia maritima              | p^D      | p^D         |              |          |          |           |                     |                      | [89], [90]       |
| Myosotis                        | 171.60^A |              |              |          |          |           |                     |                      | [81]   |
| Pelargonium spp.                | p^D      | p^D         | p^D          | p^D      | p^D      | p^D       | 12.52^s             |                      | [5], [91]        |
| Petunia                         | 53.2^2   | 31.3^2      | 49.0^4       | 2.6^4    | 87.1^3   | 8.5^3     | 28 – 114^9         |                      | [5], [92], [93]   |
| Rosa                            | 357.0    | 31.2        |              | 140.4    | 153.1    | 2.3 - 7.0^B | 71.4 - 397.4^A,9   |                      | [94], [95], [71], [96] |
| Tagetes erecta                  | 33       | 3.8         |              | 0.75^s   |          |           | 70.42^B            |                      | [5], [97]        |
| Species                  | p<sup>D</sup>  | p<sup>D</sup>  | p<sup>D</sup>  | p<sup>D</sup>  | p<sup>D</sup>  | Oxygen radical absorbance capacity |
|-------------------------|----------------|----------------|----------------|----------------|----------------|-----------------------------------|
| Tagetes patula          | 0.25<sup>1</sup> | p<sup>D</sup>  | p<sup>D</sup>  | p<sup>D</sup>  | p<sup>D</sup>  | 0.076 - 0.433                     |
| Torenia sp.             | 0.9 - 41.0<sup>9</sup> | 210.96<sup>9</sup> | 4.2 -134.9<sup>9</sup> | 5.0 - 152.7<sup>9</sup> | 8907.50<sup>9</sup> | 68[68], [99], [100], [101]        |
| Tropaeolum majus        | 4.77<sup>8</sup> | 32.208<sup>8</sup> | 32.06<sup>8</sup> | -              | -              | 68.12<sup>8</sup>                |
|                        |                |                |                |                |                | 7111- 18719<sup>8</sup>          |
| Tulipa sp.              | p<sup>D</sup>  | p<sup>D</sup>  | p<sup>D</sup>  | 3.8 - 4.0<sup>9</sup> | 29.23<sup>9</sup> | [102], [103], [104], [105]       |
| Viola cornuta           | 70.0<sup>7</sup> | 1350<sup>7</sup> | 25.0<sup>8</sup> | 8.6 - 21.8<sup>9</sup> | 8.8-14.2<sup>9</sup> | 1.2-15.9<sup>9</sup>          |
| Viola witrockiana       | 1.9 - 16.7<sup>9</sup> | 8.6 - 21.8<sup>9</sup> | 8.8-14.2<sup>9</sup> | 0.35 - 13.6<sup>9</sup> | 0.82 - 36.55<sup>9</sup> | [5], [107], [108]                |

1 the content of individual anthocyanins is given together with their derivatives like glycosides, rutinosides and others, as a sum of identified anthocyanins
2 Oxygen radical absorbance capacity
3 Presence identified but no quantitative data available
4 Average value for 8 cultivars
5 Average value for 3 cultivars
6 Average value for 4 cultivars
7 Antioxidant capacity was measured using 2, 2- diphenyl-1-picrylhydrazyl (DDPH), data expressed as percent inhibition of DPPH
8 g/100 g FW, there is sum of three cyanidins, the share of cyanidin-3-galloylsophoroside is 60-90%
9 Equivalent of pelargonidin mg per gram of sample DW
10 Equivalent of mg cy-3-glu/100 g FW and DW for Hibiscus, Petunia, Rosa, Tropaeolum,
11 Depends on the cultivar
12 Depends on the season
13 Wild species Tulipa humilis

The content of anthocyanins has not been tested yet in many edible flower species (Table 1). A number of studies were carried out in the 1980s and 1990s to find individual anthocyanins, sometimes also their derivatives [80, 81, 88, 91]; however, there were no technical possibilities to allow the measurement of anthocyanin content. The highest total amount of anthocyanins was found in petals of perennial hibiscus; annual flowers with the highest anthocyanin concentration were Dahlia sp., Petunia sp. and Tropaeolum majus [75, 92, 93]. So far, it has been found that all species grown on balconies, terraces or roofs of houses, whose flowers are edible, show high antioxidant capacity [5]. The content of anthocyanins strictly depends on the cultivar [91, 93, 109] and the development phase of flowers [109, 110]. Annual flowers grown in the human environment can be a very important source of anthocyanins [71, 77, 98], but also perennial flowers grown in larger pots, such as roses or hibiscus, are known to be an excellent source of polyphenols [33, 87, 95, 96].

6. Pharmacy in the neighbourhood (balconies, roofs, terraces)

We live in a world where only a small percentage of land remains relatively undisturbed. The urban landscape is not only to be functional, but also to actively provide cultural experiences and to create a harmonious structure [111]. Nowadays, terraces or residential courtyards in an urban or agricultural environment take up the role of kitchen gardens, contributing not only to the development of urban agriculture, but also to increasing the availability of health-promoting substances. People often turn balconies, roof terraces, or patios into an attractive space to dine and entertain with stylish lighting and furnishing ideas. At such “home plots” the cultivation of ornamental plants occupies an important place [112]. In addition to aesthetic advantages, such a location can also be a source of edible flowers, which, in addition to stunning delicacy, can perform their health-promoting functions.
The ability of anthocyanins to induce antioxidant and detoxifying enzymes has potential implications for cancer prevention and for modifying cellular oxidant status [17]. Health and therapeutic effects of anthocyanins are related to their chemical and biochemical properties, which are partially explained by their antioxidant activities. However, anthocyanins are relatively unstable and easily oxidized. They are sensitive to many factors, like temperature, UV radiation, the presence of sulfur dioxide, some ions ascorbic acid [64, 113]. Therefore, easy access to edible flowers containing anthocyanins can support the bioavailability of these compounds to consumers.

The concentration of anthocyanins in fresh fruits and vegetables can significantly drop even during only several days of storage in a cold store, as demonstrated by a study on the level of pelargonidin 3-glucoside and cyanidin 3-glucoside, the two anthocyanidin glycosides responsible for the colour of strawberries. Similar conclusions can be drawn based on various other studies, such as that on storing ‘Jonagold’ apples for 120 days, in which the amount of anthocyanins decreased during storage from 158 mg/100 g to 119 and 103 mg/100 g [115]. Growing flowers in the neighborhood and harvesting them for direct consumption just before a meal makes it possible to avoid the degradation of these beneficial compounds.

Conclusion

The existing literature indicates that many ornamental plants growing in the immediate vicinity of humans can be an abundant source of anthocyanins. Many researchers focus on widely recognized products rich in anthocyanins, such as wine or berry plants. However, more studies are needed to determine both the quantitative and qualitative anthocyanin content of edible flowers. Biotechnology offers promising methods to increase anthocyanin levels in edible flowers, whereas more widespread cultivation of flowers in containers on balconies, terraces and roofs makes it easier for humans to include them in daily diet. The literature on the subject provides sufficient evidence showing that edible flowers rich in anthocyanins may have a protective effect on human health, especially by preventing the occurrence of cancer and neurodegenerative and cardiovascular diseases.

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