Highly Sensitive ZnO NWFET Biosensor Fabricated Using Top-Down Processes

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Abstract: A highly sensitive low-doped ZnO nanowire field effect transistor (NWFET) biosensor has been fabricated and measured. The low doped biosensor with NWFET transducer was used to sense charge of the following substances: lysozyme (LYSO), phosphate buffered saline (PBS), bovine serum albumin (BSA). It achieved maximum sensitivity of -543.2 % for the PBS-LYSO protein and 13,069 % for the PBS-BSA protein. These results were achieved because the electrical measurement and characterisation was focused on the charge effect of the LYSO and BSA acting on the ZnO nanowire subthreshold region. The nano-fabrication process is stable and reproducible. The high sensitivity of the ZnO NWFET biosensor can be exploited for selective analyte detection by functionalizing the nanowire surface with antibodies and/or other biomolecular probe molecules.

Introduction

Researchers [1–10] continue to investigate zinc oxide nanowire field effect transistor (NWFET) transducer as a sensing agent for protein molecules. The task of the transducer is to convert the charge of protein molecules into electrical signal that can then be transmitted for processing [11, 12, 13]. The device has practical advantages: low costs, abundant, non-toxic, transparent, large excitonic binding energy of 60 meV, soluble, compatible with intercellular material, and wide and direct band gap of 3.37 eV making it highly sensitive [14–19]. These advantages make it a highly attractive sensor for the future semiconductor industry.

The wide band gap is highly desirable as it increases sensitivity. To further increase device sensitivity, three things are investigated: gate oxide made of Al₂O₃, low doping and the nanowire structure [1, 20] of the ZnO channel. Al₂O₃ will be used because it is compatible with ZnO and has a high-k dielectric constant. The high-k dielectric constant allows for the oxide layer thickness to be increased without sacrificing sensitivity. Low doping of the semiconductor material increases sensitivity because it is inversely proportional to device sensitivity. This is described in equation 1 [1, 20–22] below:

\[
\text{Sensitivity (S)} = \frac{\Delta G}{G_0} = \frac{2\varepsilon_r\varepsilon_0\Phi_B N(t)}{qa^2N_D \log(1 + \frac{\varepsilon_0}{a})}
\]  

(1)

The parameters of the equation are defined as follows: \(\Delta G\) is the change in conductivity of the device, \(G_0\) is the initial conductivity, \(\Phi_B\) is the surface potential, \(N(t)\) is the density of charge states at the surface, ‘a’ is a geometry parameter, \(\varepsilon_r\) is the effective dielectric constant of Al₂O₃, and \(N_D\) is the donor concentration. Thirdly, nanowires improve sensitivity because they are comparable to most biological entities and possess very large surface to volume ratios [20–25]. Due to top-down fabrication processes engineers will have control over location, orientation and dimensions of device on wafer [3, 4, 26–28].

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This paper presents a highly sensitive biosensor that uses top-down passivated ZnO NWFET as a transducer. The high sensitivity is achieved because the electrical measurements and characterisation focuses on the charge effect of the lysozyme from chicken white (LYSO: Fluka 62970) and bovine serum albumin (BSA: Sigma A3059) acting on the ZnO nanowire subthreshold region which possess a logarithmic scale. The subthreshold voltage shift due to the proteins is used: to extract the charge sensitivity, determine protein charge polarity and estimate the charge distribution.

Experiment

Fig. 1 shows a schematic diagram of the biosensor configuration. The experimental procedure for the ZnO NWFET and its application as a biosensor is as described in [5]. The recipe [5] was altered so as to improve sensitivity. The main improvement made was to the remote plasma atomic layer deposition (RPALD) process. All other steps were kept relatively the same. RPALD is capable of producing highly conformal and quality films [3]. The process is cyclic and is based on the number of reactants. For ZnO films, the cycle depends on two reactants: metallisation and oxidation. Metallisation uses diethyl zinc (DEZ) as the zinc (Zn) metal precursor. Purge and pump steps are used to separate the execution of the reactants and to remove any by-products. Before deposition: the wafer (substrate) is pre-heated to a temperature used for deposition (190 °C) and it is also cleaned with an O₂ plasma so as to remove any polymer layer. During the metallisation step, the DEZ (Zn (C₂H₅)₂) is absorbed onto the surface of the wafer, then on another step, O₂ is delivered to react with the absorbed DEZ [3].

Four main parameters were altered: film thickness, DEZ dose time, O₂ dose time and the order of gas flow. DEZ dose time was reduced from a value of 1.0 s to a value of 50 ms, while O₂ dose time was reduced from 4.0 s to 2.65 s. The doping concentration [3] in ZnO depends on oxygen defects (oxygen interstitials (Oᵢ), oxygen antisite defects (Oₛᵢ), and oxygen vacancies (Vₒ)). The recipe [5] had more Zn than O. By reducing the DEZ dose time from 1.0 s to 50 ms, it caused the O content to be much higher than the Zn content which increased the doping concentration. This means oxygen interstitials Oᵢ and oxygen antisite defects Oₛᵢ have more effect on the doping concentration than oxygen vacancies Vₒ. To reduce the interface state charge, the ZnO deposition recipe was modified. Instead of flowing O₂ first into the chamber, DEZ was flowed first. An O₂ plasma tends to damage the SiO₂ surface, thereby increasing the amount of interface state charge. The deposition of a zinc layer first provides some protection against surface damage from the O₂ plasma.

The device was used to measure three aqueous solutions: bovine serum albumin (BSA: Sigma A3059 at concentration of 0.5 µM), lysozyme from chicken white (LYSO: Fluka 62970 at concentration of 0.5 µM) and phosphate buffered saline (pure- PBS) (150 µM NaCl, 10 µM phosphate, pH 7.4), [5]. The electrical I-V characterization of the ZnO NWFETs was done using an Agilent Technologies B1500A semiconductor parametric analyser [5]. All protein detection measurements were carried out using a grounded reference electrode. Four probes were used to measure the characteristics. The probes were for the source, drain, back-gate and top-gate. The procedure for protein detection in this paper is comparable to other authors [1, 11, 13, 22, 24, 25].
Fig. 1: Schematic illustration of the fabricated biosensor.

Fig. 2 shows SEM cross-sectional image of the ZnO nanowire device. The low doped device was found to have a channel length of 20 µm, a height of 150 nm and a width of 70 nm. The width of the nanowire was measured halfway down the nanowire which is the average of the widths measured at different heights.

Fig. 2: Cross-sectional SEM images of ZnO nanowire transducer.

Procedure for protein sensing

Four different environments were measured: AIR, PBS, lysozyme from chicken white (LYSO: Fluka 62970), bovine serum albumin (BSA: Sigma A3059) [5]. Measurements were carried out in air, then in liquid and then in air again. To measure in liquid, a 5 µl droplet of the aqueous solution was introduced in the SU-8 well so as to make contact with channel. There is a wait period between measurements where the analyte solution is allowed to soak onto the surface. The low doped biosensor was measured at short time intervals of five minutes. Each environment is measured five times, once every five minutes. Table 1 shows the sequence of measuring the biosensor. The mechanism of the protein adsorption on the Al$_2$O$_3$ surface is in equilibrium with the protein concentration in solution.
Table 1: Experimental procedure that describes the sequence of measuring the low doped biosensor for four different environments (AIR, PBS, lysozyme from chicken white (LYSO: Fluka 62970), bovine serum albumin (BSA: Sigma A3059)) [5].

| Bio-Sensor       | Procedure for protein sensing                                                                 |
|------------------|-----------------------------------------------------------------------------------------------|
| Low doped device | 1. Step 1: Measure device in air = $I_{DS}$-$V_{GS}$ (AIR1), Afterwards apply PBS, 1 hour wait |
|                  | 2. Step 2: Check PBS, then measure $I_{DS}$-$V_{GS}$ (PBS1), 5 minutes wait                    |
|                  | 3. Step 3: Check PBS, then measure $I_{DS}$-$V_{GS}$ (PBS2), 5 minutes wait                    |
|                  | 4. Step 4: Check PBS, then measure $I_{DS}$-$V_{GS}$ (PBS3), 5 minutes wait                    |
|                  | 5. Step 5: Check PBS, then measure $I_{DS}$-$V_{GS}$ (PBS4), 5 minutes wait                    |
|                  | 6. Step 6: Check PBS, then measure $I_{DS}$-$V_{GS}$ (PBS5), Remove PBS and add lysozyme. 1.5 hour wait, Remove lysozyme, then add it again |
|                  | 7. Step 7: Apply LYSO, then measure $I_{DS}$-$V_{GS}$ (LYSO1), 5 minutes wait                  |
|                  | 8. Step 8: Check LYSO, then measure $I_{DS}$-$V_{GS}$ (LYSO2), 5 minutes wait                  |
|                  | 9. Step 9: Check LYSO, then measure $I_{DS}$-$V_{GS}$ (LYSO3), 5 minutes wait                  |
|                  | 10. Step 10: Check LYSO, then measure $I_{DS}$-$V_{GS}$ (LYSO4), 5 minutes wait                |
|                  | 11. Step 11: Check LYSO, then measure $I_{DS}$-$V_{GS}$ (LYSO5), Remove Lysozyme (rinse 10 times with DI water), add BSA. 1.5 hour wait, Remove BSA, then add it again |
|                  | 12. Step 12: Apply BSA, then measure $I_{DS}$-$V_{GS}$ (BSA1), 5 minutes wait                  |
|                  | 13. Step 13: Check BSA, then measure $I_{DS}$-$V_{GS}$ (BSA2), 5 minutes wait                  |
|                  | 14. Step 14: Check BSA, then measure $I_{DS}$-$V_{GS}$ (BSA3), 5 minutes wait                  |
|                  | 15. Step 15: Check BSA, then measure $I_{DS}$-$V_{GS}$ (BSA4), 5 minutes wait                  |
|                  | 16. Step 16: Check BSA, then measure $I_{DS}$-$V_{GS}$ (BSA5), Remove BSA (rinse 10 times with DI water) |
|                  | 17. After H2O dried (> 48 hours). Re-measure $I_{DS}$-$V_{GS}$ in air (AIR2)                  |

Results

Fig. 3 shows the electrical characteristics of the low doped device in air before analyte solution measurements. Fig. 3 (a) shows $I_{DS}$-$V_{DS}$ biosensor measurements in air and darkness with gate bias varying from -20 V to 0 V in steps of 2.0 V and drain bias from 0 V to 40 V. It shows clear linear, pinch-off, and saturation regions. Fig. 3 (b) is the subthreshold and linear $I_{DS}$-$V_{GS}$ plot of the NWFET in air before being exposed to the analytes. The characteristics were measured with $V_{DS}$ at 12.0 V. The device has a subthreshold slope of 170 mV/decade and a threshold voltage of -2.5 V, obtained by extrapolation of the linear characteristic.

![Fig. 3: Biosensor characteristics measured in air and darkness](image)
Fig. 4 shows the measured electrical characteristics of the biosensor in different bioanalyte solutions which were measured at $V_{DS} = 12$ V and top gate voltage (reference voltage) of 0V. Each environment within the experiment five measurements were made at time intervals of five minutes. The time interval between the different analyte solutions was one hour thirty minutes.

The five black curves (AirBefore) in Fig. 4 show the first set of electrical measurement in air. These five curves are clustered together and there is no overall trend concerning the voltage shift between the curves. Then five red curves (PBS) were measured as shown in Fig. 4 which are the second set of electrical measurement. These PBS curves are clustered closely together but tend to drift over time. They have a maximum voltage shift of 0.7 V at a fixed $I_{DS}$ of 20.0 nA. Fig. 4 (a) shows that PBS from AirBefore causes an average subthreshold shift in $V_{GS_{sense}}$ from -6.025 V to +21.099 V.

Then five blue curves (lysozyme from chicken white (LYSO: Fluka 62970)) were measured as shown in Fig. 4 which are the third set of electrical measurement. These lysozyme curves are clustered together but tend to drift over time. They have a maximum voltage shift of 0.88 V at a fixed $I_{DS}$ of 20.0 nA. Fig. 4 (a) shows that lysozyme from chicken white (LYSO: Fluka 62970) causes an average subthreshold shift in $V_{GS_{sense}}$ from 21.099 V to 17.180 V. The five brown curves (BSA) in Fig. 4 show the fourth set of electrical measurement. These bovine serum albumin (BSA: Sigma A3059) curves are clustered together but tend to drift over time. They have a maximum voltage shift of 0.7 V at a fixed $I_{DS}$ of 20.0 nA. Fig. 4 (a) shows that bovine serum albumin (BSA: Sigma A3059) from lysozyme from chicken white (LYSO: Fluka 62970) causes an average subthreshold shift in $V_{GS_{sense}}$ from 17.180 V to 23.796 V.

Finally the five green curves (AirAfter) in Fig. 4 show the fifth set of electrical measurement. These AirAfter curves are clustered together but tend to drift over time. They have a maximum voltage shift of 0.18 V at a fixed $I_{DS}$ of 20.0 nA. Fig. 4 (a) shows that AirAfter from bovine serum albumin (BSA: Sigma A3059) causes an average subthreshold shift in $V_{GS_{sense}}$ from 23.796 V to 1.080 V.

At fixed $I_{DS} = 20.0$ nA and with reference to the PBS curve, the maximum net subthreshold voltage shifts due to protein binding for lysozyme from chicken white (LYSO: Fluka 62970) and bovine serum albumin (BSA: Sigma A3059) are -4.4 V and 3.3 V respectively. The negative voltage shift of the LYSO indicates the presence of a positively charged protein and the positive voltage shift of the bovine serum albumin (BSA: Sigma A3059) signifies the presence of negatively charged protein. The ZnO NW biosensor has therefore distinguished between the positively charged lysozyme from chicken white (LYSO: Fluka 62970) and the negatively charged bovine serum albumin (BSA: Sigma A3059).
Discussion

Table 2 discusses the low doped device. The average sensitivity for sensing of lysozyme from chicken white (LYSO: Fluka 62970) protein is 417.8 % and the average sensitivity for the sensing of bovine serum albumin (BSA: Sigma A3059) protein is 5,765.8 %. These values are high when compared with literature [1–27], which means that the device is highly sensitive. The low doping concentration of $2.0 \times 10^{16}$ cm$^{-3}$ used for the NWFET channel is attributed to the high sensitivity results [1, 20–22]. This trend can be qualitatively explained by equation 1, which shows that the sensitivity is inversely proportional to the doping concentration [20].

Table 2: Summary of measurement of device sensitivity looking at voltage shift and current change between different solutions on the biosensor (low doped device).

| Low-doped Device Sensitivity | $\Delta V_{GS sense}$ (V) | $\Delta I$ (A) | % Change in Sensitivity |
|-----------------------------|---------------------------|---------------|------------------------|
| LYSO1 – PBS1                | -3.849 V                  | - 1.1705 x $10^{-6}$ A | - 293.2 % |
| LYSO2 – PBS2                | -4.024 V                  | - 1.3721 x $10^{-6}$ A | - 474.5 % |
| LYSO3 – PBS3                | -3.674 V                  | - 1.1331 x $10^{-6}$ A | - 457.0 % |
| LYSO4 – PBS4                | -3.676 V                  | - 1.2836 x $10^{-6}$ A | - 317.7 % |
| LYSO5 – PBS5                | -4.375 V                  | - 1.5294 x $10^{-6}$ A | - 543.2 % |
| BSA1 – PBS1                 | 2.802 V                   | 3.9621 x $10^{-7}$ A  | 13,069 % |
| BSA2 – PBS2                 | 2.276 V                   | 2.8133 x $10^{-7}$ A  | 3,582 % |
| BSA3 – PBS3                 | 3.326 V                   | 2.4296 x $10^{-7}$ A  | 4,859 % |
| BSA4 – PBS4                 | 2.800 V                   | 3.9447 x $10^{-7}$ A  | 4,106 % |
| BSA5 – PBS5                 | 2.276 V                   | 2.7308 x $10^{-7}$ A  | 3,213 % |

The previous fabricated device [5] achieved maximum net subthreshold voltage shift of 340 mV and 700 mV for the LYSO and BSA solutions respectively. A NWFET sensitivity of 72 % was achieved for the LYSO proteins while the BSA proteins resulted had a sensitivity of 98%. These values are insignificant compared with those currently achieved. Comparing the low-doped device with the state-of-the-art devices shows that the device is more sensitive. Silicon is still the most used material for nanowire biosensors. The device made by X. Duan, et al., [29] shows the highest and stable sensitivity recorded in terms of ‘voltage shift’ having a value of 320 mV. A. Fulati, et al., [30] recorded a much lower value of voltage shift of 65.2 mV. The range of voltage sensitivity recorded by most researchers [1–43] for biosensors is between 50 and 400 mV. This paper records greater voltage shift ($V_{GS sense}$) which makes the device more sensitive. The low doped device is more sensitive than X. Duan, et al., [29] by a value of 4.06 V. The sensitivity is even much higher when compared with A. Fulati, et al., [30].

It must be noted that the charge-based measurements carried out are non-specific. The aim is to analyze the charge effect but in future the device will be functionalised to sense particular substances such as ions, proteins, DNA, and/or viruses. Antibody-functionalized nanowire sensors can be used to detect the proteins such as interleukin-8 (IL-8) and tumor necrosis factor-alpha (TNF-α). Antibody–antigen dissociation constants must be in good agreement with low-salt enzyme-linked immunosorbent assays (ELISAs).
Conclusion

A highly sensitive low-doped ZnO NWFET biosensor has been fabricated and measured. The low doped device had maximum sensitivity of -543.2 % for the PBS-LYSO protein and 13,069 % for the PBS-BSA protein. The device response is stable and reproducible. The high sensitivity of the ZnO NWFET biosensor can be exploited for selective analyte detection by functionalizing the nanowire surface with antibodies or other biomolecular probe molecules.

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