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Green Chemicals from Used Cooking Oils: Trends, Challenges and Opportunities

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ABSTRACT
Food waste reduction is fundamental for sustainable development and pursuing this goal, recycling and the valorization of used cooking oil (UCO) can play a major contribution. Although it has been traditionally used for biofuels production, the oleochemical potential of UCOs is vast. UCOs can be used as feedstock for a large variety of value added green chemicals including plasticizers, binders, epoxides, surfactants, lubricants, polymers, biomaterials, and different building blocks. Thus, UCOs transformation into functional chemicals can bring long-term stability to the supply chain, avoiding the current dependence on commodity products. In this regard, this work describes some of the potential benefits of using UCOs as feedstock in oleochemical biorefineries. Also, some of the most recent investigations on the valorization of UCOs other than biofuel are presented. Finally, major challenges and future directions are discussed.

Keywords. Used cooking oil, biobased chemicals, value-added products, challenges

1. Introduction

Reduction of food loss and waste is paramount to fulfil the UN’s Sustainable Development Goals, and crucial to curtail their associated economic, social and environmental life cycle impacts. Current estimations indicate that average per capita food waste generation in Europe ranges between 173 - 290 kg/person-yr [1]. Equally alarming numbers (in kg/person-yr) are observed in Australia (361), USA (278), Canada (123), India (51), China (44), and other countries [2]. Reported data also reveals that nearly 60% of food waste is generated as consumption and post-consumption residues (e.g. bones, cooking oils, peels, leftovers, etc.), and a large fraction is unavoidable or inedible [1]. In order to mitigate the impacts, different circular economy approaches have been proposed for the exploitation and valorization of food waste via transformation into a large variety of chemicals, materials and fuels through a biorefinery
approach [3-6]. The potential valorization processes and the targeted products largely depend on the nature and composition of waste. Besides carbohydrates, starches, and proteins, a large fraction of typical food residues corresponds to fats and oils (i.e. up to 25% wt. on a dry basis, [4, 7]). Particularly, among the different food wastes, discarded used cooking oil (UCO) is a major source of lipids. While suitable processes are required to extract the lipid content from most food waste [7], lipids in UCOs are readily reachable. This explains the existing UCOs collection chains and the different processes for their valorization at an industrial scale [8-10].

UCOs are mainly generated in households and hospitality sectors (HORECA – Hotels, Restaurants, Casino-Cafe-Catering) [11, 12], and the current global production is estimated between 20 and 32% of total vegetable oil consumption (41 - 52 Mt/yr., [9, 13]). This broad range is a result of the different culinary customs and consumption trends in the different regions, which also play a major role in the nature, chemical composition, and content of impurities in the UCOs [8]. Because of unconscious behaviors, absence of regulations, or lack of law enforcement, most UCOs are generally disposed through sinks and syphons, or within the solid residues that are sent to landfills. In addition to ecosystems pollution and public health impacts, this mismanagement generates a variety of cascading problems including sewage clogs, wastewater overflow, costly damage to infrastructure, vectors and pests, nauseous odors, higher operating costs at central wastewater treatment plants, etc. In order to mitigate all these problems, a fraction of UCOs have been typically recovered and reused as oleochemical feedstock, mainly for the production of low added-value commodities such as biofuels (e.g. biodiesel, hydrogenated vegetable oil - HVO), soaps and animal feed. This has created a small but solid market of nearly 600 Million USD/yr., growing at an average annual rate of 4% [14], and with prices in the range from 620-865 USD/t during the last 12 months [15]. In spite of the growing trend, and because of the low added-value of current derivatives, UCOs market is highly vulnerable to the change of economic and political environment. This vulnerability could be reduced by diversifying the portfolio including high value-added byproducts, thus bringing long-term stability into the entire valorization chain. In this regard, this work describes most current research on the transformation of UCOs into high value added biobased products, the challenges for a successful industrial implementation, the associated benefits, and some future directions.

2. Potential benefits of UCOs as Oleochemical feedstock

The main raw materials of the oleochemical industry are vegetable oils and animal fats, with the former having a 99.9% share in volume, and the remaining small fraction corresponding to butter, fish oils, and
fats from animal rendering [16]. Figure 1 presents the historical production of vegetable oils and the corresponding distribution regarding final use. As observed, 68% of current world production is used for food applications (i.e. cooking oils, food ingredients), and 23% in biofuels, mainly biodiesel (~1kg\text{Oil}/1kg\text{Biodiesel}). The remaining fraction (~9%) is destined for feed and other oleochemical uses including drop-in applications (e.g. additive for polymers, resins, asphalt, lubricants, greases, drying oils, rubber products, etc.) and as feedstock for different chemical derivatives. Taking into account the estimated global UCO generation (41 Mt, [9]), this amount can replace the virgin vegetable oil currently required as feedstock for the oleochemical industry, a part of which is used as a biodiesel feedstock. Hence, the exploitation of UCOs as chemical feedstock within a circular economy model would help to reduce the environmental and social impacts associated to both, the edible oil and the oleochemical industries.

**Figure 1.** Historical world production of vegetable oils and share among final uses (Data from [16]).

Figure 2 schematically presents the different stages in which negative social and environmental lifecycle impacts could be mitigated by using UCOs as oleochemical feedstock [17-22]. This is based on a 100% efficiency in UCO collection and reuse, and considering the more conservative estimation of UCOs generation with respect to vegetable oil consumption (20%, [9]). Also, it is important to consider that for
instance in the EU, only 45-50% of collectable UCO is recovered, with a best-case scenario of 70% [11, 23].

**Figure 2.** Potential impacts mitigation of employing a circular economy approach in the exploitation of UCOs as oleochemical (OChe) feedstock considering lifecycle stages (Cradle-to-gate.). Functional unit: 1 t vegetable oil. Assuming a 90% yield of refined UCO from the collected one.
3. **High value added application for UCOs**

The valorization of UCOs via transformation into suitable oleochemical products have captured the attention from academic and industrial researchers during the last two decades. Figure 3 presents the evolution of the scientific production (i.e. papers and patents) dealing with the exploitation of UCOs. While the studies on biodiesel are still predominant, there is an increasing trend to explore novel applications, mainly focused on value-added products. In addition to the availability of financial resources, most research have been conducted in countries where UCOs mismanagement can be a major problem, either because of their large population (e.g. China, India), or for the large per capita generation (e.g. USA, Indonesia, S. Korea). Most EU countries are grouped as “others”, and in this case, their large scientific productivity has been promoted by the public policies of the community [20].

![Figure 3. Number of publications and patents on the valorization of used cooking oil in last decades, and share by major contributing countries (March, 2020). Searching terms: TITLE-ABS-KEY ("Used cooking oil" OR "Waste cooking oil" OR "Yellow grease" OR "Brown grease" OR "Trap grease")](Source: [24-26]).

Recent reports indicate that biobased products can have a large market growth in the coming years, if similar policies and subventions to those implemented to the production of biofuels, are also implemented for green chemicals [27]. In this context, a variety of new processes and products have been developed for UCOs valorization, evolving from basic drop-in applications, to more complex thermochemical, chemical and biochemical transformations [28-31]. More recently, further exploration has enabled the development of value added products from the crude glycerol obtained as co-product from UCOs-based biodiesel processes [32, 33]. While this is not intended to be a comprehensive review of the available literature, Table 1 presents a summary of the most recent attempts for UCOs harnessing, including the production of plasticizers, binders, epoxides, surfactants, lubricants, polymers, biomaterials, building blocks etc.
**Table 1.** Most current attempts on the production of biobased chemicals from UCOs and UCOs-based glycerol

| Application       | Process                                                                 | Product                                                                 | Chemistry behind product use                                                                 | Highlights                                                                 | Ref.        |
|-------------------|------------------------------------------------------------------------|------------------------------------------------------------------------|----------------------------------------------------------------------------------------------|---------------------------------------------------------------------------|-------------|
| UCOs Valorization | Transesterification of UCOs biodiesel with 2-ethylhexanol and further epoxidation | Epoxidized 2-ethylhexyl fatty ester                                      | Oxirane reacts with compounds containing active hydrogen atoms (e.g. water, organic acids, alcohols, Halides) | Ep-WCOEtHEs in PVC enhanced the overall mechanical property and thermal stability, with no significant change in migration-resistant performance. | [34]        |
|                   | Epoxidation                                                             | Epoxidized UCO                                                          | Enhanced thermo-oxidative stability by reducing unsaturations                                | Primary plasticizer for PVC films, without the need of other additives, resulting in samples with good thermal stability and mechanical properties | [35]        |
|                   | Transesterification of UCO with methanol and epoxidation of methyl ester. Esterification with citric acid, and final acetylation with acetic anhydride | Acetylated FAME citric acid ester (Ac-FAME-CAE)                         | Hydrogen bonding of the 8 carboxylic groups of Ac-FAME-CAE with PVC polymer to enhance thermal stability | Similar plasticizing performance to DOP                                    | [36]        |
|                   | Epoxidation                                                             | Epoxidized UCO                                                          | Oxirane reacts with compounds containing active hydrogen atoms (e.g. water, organic acids, alcohols, Halides) | Homogeneous and heterogeneous catalysts were tested. Dimers and oligomers formed using H_2SO_4 as catalyst | [37]        |
|                   | Esterification and transesterification with methanol and amino-methylation (Mannich reaction) | Mannich base of UCO biodiesel                                          | chlorine atoms of PVC substituted with Mannich base of UCO biodiesel                          | Lower thermal stability due to the content of active secondary amine group | [38]        |
| Asphalt/pavement binder | Drop-in. 5%wt. addition                                                  | Asphalt binder with light components from UCO                          | Carbonyl groups reacting with binders                                                        | Low temperature crack resistance and softness of the asphalt binder are improved | [39]        |
|                   | Drop-in. 0.4 - 0.8 %wt. addition                                        | Macadam pavement                                                       | UCO physically cover aggregates                                                              | Improves cracking and fatigue resistance                                     | [40]        |
|                   | Co-pyrolysis of UCO with rubber                                         | Rubber/UCO binder                                                      | UCO reacts with rubber polymers during de-sulfurization and pyrolysis                        | Improved low temperature properties of binder Improved rheological properties of asphalt | [41, 42]   |
|                   | Drop-in. 5%wt. addition                                                 | Binder replacement                                                      | Unsaturations bond with asphalt macromolecules and binder                                    | Treated waste cooking oil can be used as a replacement of asphalt binder in asphalt mixtures | [43, 44]   |
|                   | Drop-in. 5% UCO addition                                                | Asphalt binder                                                         | Rheological modifier of asphalt binder                                                       | Addition of waste cooking oil as binder replacement was improve the durability performance of asphalt mixture. | [45]        |
| Masonry binder    | Drop-in. 10%wt. addition                                                | Construction block                                                     | Oxy-polymerization and crosslinking                                                          | WasteVege block does not require the use of any form of cementitious or pozzolanic materials or water. | [46]        |
| **Epoxidized biodiesel** | Enzymatic transesterification and epoxidation | Epoxidized UCO biodiesel | Oxirane reacts with compounds containing active hydrogen atoms (e.g. water, organic acids, alcohols, Halides) | Impurities had a negative effect in the epoxidation. Oxirane value 2.5. |
|---|---|---|---|---|
| **Polyol / Polyurethane** | Epoxidation of UCO and hydroxylation with diethylene glycol | UCO-based polyl | Hydroxyl groups react with isocyanate to form urethane bonds | The application of sulfuric acid in this experiment required a much higher temperature than in the case of the catalysts based on tetrafluoroboric acid and a longer reaction time. Satisfactory polyurethane bio-foams can be produced by replacing 40-60% of polyl with biobased alternative. |
| **Lubricant** | Epoxidation and hydroxylation with methanol, ethanol, and 2-ethyl hexanol, esterification with hexanoic anhydride | UCO and UCO FAME poliol hexanoic ester | Enhanced thermo-oxidative stability by reducing unsaturations and hydroxyl groups | Products are compliant to standard lubricant specifications in terms of viscosity, viscosity index and pour point; with much higher biodegradability. |
| **Surfactant** | Epoxidation of UCO | Epoxidized UCO | Enhanced thermo-oxidative stability by reducing unsaturations | Epoxidized UCO exhibit highly desirable and enhanced physicochemical properties in all the aspects for environmentally friendly biolubricants. |
| | Enzymatic hydrolysis and esterification | Fatty acid neopentyl glycol ester | Enhance lubricity taking into account viscosity profile | A maximum conversion of 94% was found after 24 h using immobilized enzyme. |
| | Drop-in. 15%wt. addition | UCO dispersible Cu nanoparticles | Unsaturations complexed on the nanoparticles surface. | Particles are synthesized in UCO, and directly used as additive. The formulations are stable, without segregation even after months. |
| **Liquid detergent** | Transesterification with methanol, sulfonation of methyl ester, and neutralization with NaOH | Methyl ester sodium sulfonate | Sulfonation provides polar moieties to FAME turning it into surfactant | Yield of methyl ester sulfonic acid (MESA) after sulfonation was obtained 77.20%. Methanol reduced substitution of methyl groups into a disalt. |
| **Biopolymer precursors** | Saponification of UCO with KOH, Acidification to FA, esterification with methanol to FAME, reduction to Fatty alcohol, esterification with chloroacetic acid, and amination | Diamininium chloride gemini-surfactant | Negatively charged carboxylic oxygen and chloride from surfactant are adsorbed on metal surface creating a protecting barrier layer | Efficacious inhibitor for steel (N80) corrosion. |
| | Transesterification with methanol and sulfonation of methyl ester | Methyl ester sulfonate | Sulfonation provides polar moieties to FAME turning it into surfactant | Liquid detergent comprising of 15% MES concentration and 0.1% ZnO nanoparticles. |
| | Transesterification and ethenolysis | Ethenolyzed and self-metathesized products | Olefinic bonds in reaction products can be used for further polymerization | A novel renewable lipidic source of spent hen for ethenolysis is exploited for the first time. |
| Biobased Polymers | Epoxidation, hydroxylation with water, polymerization with Methylene diphenyl diisocyanate | UCO-based polyurethane doped with lithium iodide | Hydroxyl groups react with isocyanate to form urethane bonds | UCO-based PU could be used as a potential host for polymer electrolyte | [59] |
|------------------|----------------------------------------------------------------------------------|-----------------------------------------------|-------------------------------------------------|-------------------------------------------------|-------|
| Fermentation     | Polyhydroxyalkanoate and astaxanthin-rich carotenoids                           | Polyhydroxybutyrate [P(3HB)]                  | Biodegradable, elastomeric, thermoplastic and biocompatible polymer | 1% v/v UCO was used. 1 g/L of PHA               | [60] |
| Fermentation     | Polylpolyhydroxyalkanoates (PHAs) - (R)-3 hydroxyoctanoic acid and (R)-3-hydroxydecanoic acid monomers |                                 |                                                                 | WCO could provide better accumulation of P(3HB) in C. necator H16 compared to other common plant-based oil. The higher production of P(3HB) was approximately 80 wt%. | [61] |
| Fermentation     |                                                                                | Polyhydroxyalkanoates (PHAs) - (R)-3 hydroxyoctanoic acid and (R)-3-hydroxydecanoic acid monomers | Low molecular weight 18342 kDa. Low yields probably caused by inhibitory compounds in UCO |                                               | [62] |
| Fermentation     | Drop-in. 2.3%wt. addition                                                       | Microbial oil                                 | Carbon source for biomass | Mixture of UCO with crude glycerol enhances oil accumulation in yeast | [63] |
| Fermentation     | Lipase                                                                           | Carbon source for biomass                    |                                                                 | maximum lipase activity (12 000U/L), also lipid-rich biomass (48% of lipids mass per dry cellular mass), | [64] |
| Fermentation     | d- and l-Limonene                                                                | Carbon source for biomass                    |                                                                 | UCO is superior substrate than glucose, but low titers obtained 2.5-2.7 mg/L | [65] |
| Fermentation supplement |                                                                                   |                                               |                                                                 |                                                                                      |       |
| 3D printing resin | Acrylation                                                                       | Triacylglycerol acrylate                      | UV-promoted crosslinking of acrylic moieties in acylated UCO | Higher biodegradability of printed plastics, no photo inhibitors required | [67] |
| Emulsion liquid membrane | Drop-in. 50-80%wt. addition                                                      | Emulsion                                      | UCO is used as organic solvent of liquid emulsion | 99.1 efficiency in the recovery of organic dyes from contaminated water | [68] |
| Flotation oil     | Pyrolysis                                                                         | Deoxygenated hydrocarbons                    | Adsorption of hydrocarbons on coal surface        | UCO pyrolysis products possessed strong collecting ability and better selectivity, and can replace diesel as a coal flotation collector. | [69] |
| Bioadsorbent      | Impregnation and pyrolysis                                                       | Ordered micro-mesoporous carbon nanocasted on HZSM-5/SBA-15 | Carbonaceous material contains oxygen-rich groups suitable for adsorption of cationic dyes. Ordered micro-mesoporous structure combines size selectivity and high diffusion rates | Material exhibits high adsorption capacity comparable to activated carbons | [70] |
| Carbon source | Fermentation | Biobased building blocks |
|---------------|--------------|-------------------------|
| Hydrogen      | Bioconversion of crude glycerol by sub-tropical mixed and pure cultures. 15.14 mol L H/mol glycerol | [71] |
| Lactic acid   | Bioconversion to lactic acid by *Rhizopus microsporus*. Lactic acid average production of 3.99 g/L | [72] |
| 1,3 Propanediol | The effect of crude glycerol impurities on 1,3-propanediol biosynthesis by *Klebsiella pneumoniae DSMZ 2026*. 9.69 g/L 1,3-PD (yield 0.21 g/g, productivity 0.80 g/L/h) was obtained after 12 h | [73] |
| 1,3 Propanediol | Production to 1,3-Propanediol and Lactate by a Microbial Consortium. Impurities in GWCO did pose a great challenge to microbial growth and metabolism. In fed batch fermentation, 27.77 g/L 1,3-PDO and 14.68 g/L LA were achieved. | [74] |
| Valeric acid  | Anaerobic fermentation with open microbiome. High valerate extraction rates with medium and maximum values of 12.9 and 30.0 g COD/m3 day, were obtained with low ethanol addition (15% of the glycerol-COD) | [75] |
| 1,3 Propanediol | 1,3-PDO production with a mixed culture, A maximum productivity of 7.49 g/Ld | [76] |
| Lipids        | Lipid production via fermentation with *Trichosporon oleaginosus*. The highest lipid yield 0.19 g/g glycerol was obtained at 50 g/L purified glycerol in which the biomass concentration and lipid content were 10.75 g/L and 47% w/w, respectively. | [77] |
| Citric acid, Succinic acid | A Suitable Substrate for the Growth of *Candida zeylanoides* Yeast Strain ATCC 20367. Biosynthesis of organic acids (e.g., citric 0.66 g/L; and succinic, 0.6 g/L) was significantly lower compared to pure glycerol and glucose used as main carbon sources. | [78] |
4. Challenges and future directions

As observed, UCOs can be used as raw material for a large variety of green chemicals. In addition to the technical limitations observed in some of the processes, there are a number of issues that need to be overcome in order to enable industrial implementation. As in any other biorefinery, the supply chain plays a major role in the sustainability of the proposed production schemes. In this case, a major fraction of the generated UCOs comes from the Household segment, for which very low recovery efficiencies are typical (< 6%, [11]). Hence, it is necessary to deploy effective policies and regulated practices to enhance UCOs recycling and collection rates, under multi-stakeholders considerations (i.e. authorities, generators, collecting companies, biorefineries). This also indicates that there is need to optimize the collection schemes to ensure that the consumed resources (e.g. energy, financial) do not surpass those from the obtained UCOs, mainly when the biorefineries operate as centralized facilities far from the source. For instance, one study has shown that from a life cycle perspective, biodiesel from European rapeseed UCOs is less sustainable than petroleum diesel and even less than biodiesel from Indonesian’s palm oil UCOs [18].

Another major challenge is the highly heterogeneous nature of UCOs. They exhibit a large variability in physicochemical and sensory properties, and a substantial amount of impurities [8, 79, 80], resulting from different diets, culinary practices, and management procedures. In most of the studied processes of Table 1 there were reports of impurities in the raw material affecting the catalytic and biologically conversion steps, the drop-in uses, and even the thermochemical transformations. Also, unpleasant sensory properties (e.g. color, appearance, odour) are of major concern. Therefore, suitable upgrading processes must be implemented to enable the efficient transformation of UCOs to the desired biobased chemicals without compromising the economic feasibility [81]. Also, resilient and intensified technologies must be developed to enable the use of other types of waste lipids (e.g. trap grease, rancid oils, food/solid waste lipids, etc.). In any case, even after pretreatment and upgrading, the presence of trace impurities also might prevent that some of the derivatives could be used in sensitive applications (e.g. personal care products, cosmetics, food or pharmaceuticals) where the market is more attractive. Alternatively, they could be directed to other markets such as construction materials, asphalt, rubbers, lubricants, surfactants, fuels, etc.

A current threat to the industrial implementation of UCOs-based chemicals, is that they are strongly linked to the biodiesel market, and there are some concerns about the sustainability of this fuel especially given the rapid move from liquid fuel to electric-powered vehicles. Nowadays, UCOs biodiesel is promoted via public policies such as the Renewable Energy Directive (RED II) from EU. According to this directive,
UCO biodiesel can be double-counted, so its price can be higher than first generation biodiesel, encouraging supply and demand. Nevertheless, recent claims indicate that at least one-third of UCO-based biodiesel in the European market is fraudulent, because apparently it corresponds to the recently banned palm oil biodiesel [82]. Besides, some of the unsustainable palm oil that was prohibited in the EU has been diverted to China for animal feed in order to replace the UCOs that are currently exported to Europe [83]. These type of problems might push for revisions of RED II, which will directly affect UCOs supply for the oleochemical industry [84]. Finally, current COVID-19 pandemic is putting pressure on UCOs global trading, affecting supply, dropping prices, and reducing the potential profitability of the biorefineries [85].

5. Concluding Remarks

Used Cooking Oil is a valuable food waste that can be transformed into a large variety of products. While the use as biofuel feedstock enabled the creation of a global collection and supply chain of UCOs, only the incorporation of high value added green chemicals within the biorefineries would ensure their long-term sustainability. As presented, UCOs exploitation as oleochemical feedstock can involve large reductions in life cycle impacts, cutting the need for virgin vegetable oil, and alleviating the impacts of the current mismanaging practices for disposal. Also, by using UCO derivatives as ingredients in different end products, there is a contribution to “green” other sectors such as polymers, asphalts, cementing materials, detergents, lubricants, etc. Despite such a circular economy model around UCO seeming attractive, major challenges have to be overcome. Future developments will be mostly focused on dealing with UCOs heterogeneity and impurities content, upgrading processes, enhancing household collection, and implementing resilient and intensified processes capable of incorporating different types of waste lipids (e.g. trap grease, rancid oils, solid waste lipids, etc.).

Conflict of interest statement

Nothing declared.

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World production of vegetable oils (Mt/yr.)

- Used as Food
- Used for Biofuel
- Others

Figure
OILS/FATS INDUSTRY - CURRENT IMPACTS

1 t - Vegetable oil

0.68 t  0.23 t  0.09 t

FOOD  BIOFUEL  OChe

UCOs

0.20 t

Impacts associated with:
- Plantation
- Harvesting
- Transportation of fruits/seeds to extraction plants
- Oil extraction
- Oil Refining
- Transportation to final user

Impacts associated with UCOs mismanagement:
- Cascaded impacts from disposal in sewage and within solid residues
- Transportation to landfill
- Waste disposal
- Potential illegal use as edible oil or animal feed

OILS/FATS INDUSTRY - UCOs EXPLOITATION SCENARIO IMPACTS

0.85 t - Vegetable oil

0.17 t

UCOs

0.15 t  0.02 t

0.68 t  0.17 t  0.06 t  0.09 t

FOOD  BIOFUEL  OChe

Reduced impacts with respect to current status

Impacts associated with:
- Collection and transportation to refining facilities
- Pre-treatment
- Refining to suitable oleochemical feedstock
- Transportation to final user
- Waste disposal
HIGHLIGHTS

• World production of used cooking oils (UCOs) and current market data
• Identification of life cycle impacts reduction by exploitation of UCOs
• Review of current state of art on used cooking oil valorisation into green chemicals
• Major challenges on UCOs utilization as oleochemical feedstock