The use of the fluidized bed boiler for the disposal of the multi-material packaging waste

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INTRODUCTION

Increase in a society's standard of living is inextricably linked with an elevated and increasingly easier access to consumer goods, including food. The expected standards in this area include, among others, access to fresh drinks, including milk products, in the quantities suited to the needs of individual customers. The appearance in the 1930s of the milk packed in cardboard boxes, in the U.S. was a response to these expectations and the requirements of an easy and cheap form of transportation of such products. Years of research in these packages have led to the assumption of the standard multi-layer packaging, universally identified, in the public awareness, with the products of Tetra Pak or Combibloc.

Each of the layers of the packaging has its own function, and the whole unit constitutes a safe, cheap and easy to use package. The cardboard layer representing approximately 75% of its mass provides resistance to mechanical damage and stability of shape, which is an essential factor during transport. The layer of polyethylene or polypropylene (approx. 20% by weight of the packaging) constitutes a barrier to moisture and fat. The aluminum foil (about 5% by weight of the packaging) effectively protects the contents from penetration by air, foreign odors, light and microorganisms. This type of packages represents the exclusively non-reusable packaging. They are produced in a way which makes them impossible to recycle. Although such solutions exist, recycling of the material is technologically difficult, energy-intensive and too costly. Due to packages' high calorific value – over 20 MJ/kg, it would seem economically and ecologically justified to employ energy recycling. Under current Sustainable Development Strategy the raw materials that have been used once for production should be recovered. This would reduce the stream of waste disposed of by depositing in landfills.

In the case of the multi-material packaging one option would be to stipulate to recovery of all the raw materials, but such technology might prove to be an uneconomic solution. In the process that we have analyzed the cardboard and plastics have the chemical energy that can be recovered, in the form of heat, in the combustion process. Aluminum is recovered as a raw material. An effort is made so that at least some part of it was in an un-oxidized form.

EXPERIMENTAL PART AND RESULTS

The paper presents the results of the disposal of packaging waste from two companies: Tetra Pak and Combibloc, carried out in a fluidized bed boiler of rated thermal power 0.5 MW. The material introduced into the fluidized bed boiler underwent thermal and mechanical degradation in a sand bed of the temperature between 750 and 850°C. The process proceeds auto-thermally, without the need of additional fuel. The appropriately chosen fluidization parameters caused the separation of the solid products of combustion from the deposit material. Presence of aluminum, part of it in an un-oxidized form, was confirmed in separated dust. The gaseous products of combustion contained the traces of oxides of nitrogen and sulfur, mainly originating from the remnants of food products contained in the packaging. However, the concentration of these oxides met the requirements of emission standards.

Keywords: packaging waste, thermal disposal, fluidization, heat recovery and aluminum recovery.
with a large stream of secondary air. Oxygen content in flue gases varied in the range 12 – 18% by volume and the concentration of CO was 1200 – 2400 mg/m$^3$ (Fig. 2 and 3). During the process, observed significant drop in oxygen concentration in the exhaust gases and rise in CO content was the result of a momentary increase in the amount of the fed multi-material packaging. There are also registered the changes in the concentration of volatile organic compounds, analogous to the changes in the concentration of CO. These are outlined in Figure 3, expressed as total organic carbon (TOC).

The content of NO$\textsubscript{x}$ in the exhaust gases (Fig. 4) is (133 ± 37) mgNO$\textsubscript{2}$/m$^3$ and does not exceed the limit of 8 and, to an inconsiderable degree, depends on the changes of the stream of the fed material. In addition, the concentration of SO$\textsubscript{2}$ in the Flue gases is (131 ± 42) mg/m$^3$ and this level is lower than the limit (Fig. 4). In the absence of significant amounts of nitrogen and sulfur in materials constituting the multi-material packaging, the concentrations of NO$\textsubscript{x}$ and SO$\textsubscript{2}$ obtained are, to a larger extent, the result of the combustion of the remnants of the liquids contained in the packaging.

The material obtained from the settling chamber and cyclones (C and D in Figure 1) was subjected to granulometric analysis (test sieving was used), the results are shown in Figure 5. The material introduced to the bed underwent thermal and mechanical degradation, thus the distribution of particle sizes that deviate from the normal. Approximately 90% of the dust mass contains fractions of the diameter less than 0.43 mm. The larger grains consisted mainly of firing up the boiler sawdust blown away during the ignition of the boiler as well as aluminum flakes that did not undergo greater fragmentation in the bed during the combustion. The whole dust also contains the crushed sand from the bed. The average ratio of SiO$\textsubscript{2}$ by weight of the total contents of cyclones residues exceeded 36%, and exceeded 63% of total contents of settling chamber. Masses of aluminum in these materials were 11.0% and 61.6% (72.6% as sum) of Al included in the fuel combusted during test.

Dust concentration measurement after cyclones persisted for 28 min and was launched several minutes after the continuous dosage of the tested material was initiated. The physical parameters of the flue gases at the measuring cross-section and in the dust meter's sample gas train were collected in Table 1. The dust concentration under nor-
polished conditions (pressure 1013 hPa, temperature 273 K, dry gas) was 2.23 g/m$^3$, and the flue gases flow (the same conditions) was 1524 m$^3$/h. The dust mass flow was 3.4 kg/h. With the assumption that Al content in the dust after cyclones is the same as in the finest grade (0.00 – 0.15 mm) of the dust collected, the amount of this element emitted was estimated as 19.7% of Al transferred to the boiler. It means that not only dust chamber and cyclones but also bag filters are necessary to rise the amount of Al collection efficiency from about 70% to higher than 90%.

CONCLUSIONS

The conducted research studies point to the possibility of using the multi-material packaging waste as alternative fuel for heat production with a simultaneous recovery of aluminum. Such combination of two recovery processes allows for waste disposal with a simultaneous reduction in the waste stream discharged to the landfills, and thus its safe neutralization. Conducting the process of thermal processing in the fluidized bed boiler facilitated compliance with emission standards without an employment of any additional methods for the removal of NO$_X$ and SO$_2$ from exhaust gases. This makes it possible to limit the costs associated with the use of additional resources and facilities for air protection. On the other hand, useful products are obtained in the form of heat energy and part of the recovered aluminum, what is more, the remaining solid products are compensated and environmentally safe – dust and ash that might also be of further use, for example, as ballast for mines or construction of road embankments.

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Table 1. The physical parameters of the flue gases at the measuring cross-section and in the dust meter's sample gas train

| Parameter                      | Unit | Value at the measuring cross-section | Value in the dust meter's sample gas train |
|--------------------------------|------|--------------------------------------|-------------------------------------------|
| Temperature                    | °C   | 150                                  | 18                                        |
| Atmospheric pressure           | kPa  | 99.3                                 | 99.3                                      |
| Static pressure                | kPa  | -0.1                                 | -26.1                                     |
| Dynamic pressure               | Pa   | 74                                   | 77                                        |
| Moisture content (dry gas basis)| kg/kg| 0.080                                | 0.003                                     |
| Wet gas density                | kg/m$^3$ | 0.798                        | 0.895                                     |
| Wet gas density (normalized conditions) | kg/m$^3$ | 1.262                        | 1.320                                     |
| Dry gas density (normalized conditions) | kg/m$^3$ | 1.322                        | 1.322                                     |