Modeling of Negative Autoregulated Genetic Networks in Single Cells

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Abstract

We discuss recent developments in the modeling of negative autoregulated genetic networks. In particular, we consider the temporal evolution of the population of mRNA and proteins in simple networks using rate equations. In the limit of low copy numbers fluctuation effects become significant and more adequate modeling is then achieved using the master equation formalism. The analogy between regulatory gene networks and chemical reaction networks on dust grains in the interstellar medium is discussed. The analysis and simulation of complex reaction networks are also considered.

Key words: genetic networks, repression, master equation

1 Introduction

Recent advances in molecular biology techniques for the engineering of synthetic networks have made possible the measurement of populations of mRNA’s and proteins in simple genetic networks. Measurements of the average protein content of cells and their time dependence enabled to quantify the behavior of genetic networks [Kalir et al., 2001]. These measurements have been modeled using rate equations, mainly under quasi steady state conditions. However, real biological systems are likely be away from steady state [Smith, 1968, Murray, 1989]. Furthermore, many components of cells appear in low copy numbers and are therefore subjected to large fluctuations. Recently, such fluctuations at the level of a single cell were measured experimentally using the green fluorescent protein (GFP) [Elowitz et al., 2002, Swain et al., 2002, Paulsson, 2004]. Measurements of protein levels in single cells revealed distributions that depend on the topology of the regulatory network controlling the particular protein. For example, it was shown that negative autoregulated networks reduce fluctuations [Becskei and Serrano, 2000]. The modeling of these fluctuations cannot be done using rate equations and requires the master equation...
In this paper we consider the modeling of negative autoregulated genetic networks in cell populations and in single cells. We focus on the simplest network in which a single protein serves as a repressor for the production of its own mRNA. Such network may serve as a module or “network motif” in complex regulatory networks (Milo et al., 2002, 2004). We describe the time dependence of the system using rate equations. In commonly used models it is assumed that the population of the bound repressor proteins is in quasi steady state. We consider the dynamics of the network when this assumption does not hold. We show that in such cases the commonly used models underestimate the response of the system to variations in the external conditions. In such cases one should take into account the bound repressors as a separate population. In the limit of low copy numbers of the mRNA’s and proteins stochastic noise becomes significant. We show that in this limit the rate equations should be replaced by a master equation. The rate and master equations used in the analysis of genetic networks are closely related to those that describe chemical reaction networks on small grains. In this context, the limit of low copy number is achieved for reaction networks on interstellar dust grains, due to the sub-micron size of the grains and the extremely low flux due to the low density of the interstellar gas. This analogy is discussed and results obtained for grain chemistry, which may also be useful for genetic network analysis, are presented.

The paper is organized as follows: in Sec. 2 we consider the dynamic behavior of a simple genetic network in a cell population using rate equations. In Sec. 3 we consider the limit in which each cell contains a small population of proteins, where the stochastic features become significant. The master equation for this system is presented. The analogy between genetic networks and grain-surface chemistry is discussed in Sec. 4. A summary is presented in Sec. 5.

2 Rate equations

In genetic autoregulatory circuits the production rate of a certain product protein $A$ depends on its population size, $[A]$ (given by the average number of such proteins in a cell). In negative autoregulation, increasing the population size $[A]$ decreases the rate of production. This mechanism is commonly approximated (Rosenfeld et al., 2002, Paulsson, 2002) by the Hill function

$$ g(A) = \frac{g_{\text{max}}}{1 + k[A]} $$  \hspace{1cm} (1)
where \( g(A) \) is the production rate of \( A \) proteins, \( g_{\text{max}} \) is the maximal production (achieved in conditions where \([A] = 0\)) and \( k \) is an affinity constant. This approximation is in agreement with experiments done at steady state [Yagil and Yagil, 1971; Yagil, 1975]. Here we consider the following circuit: a population size \([R]\) of mRNA’s is produced with a maximal rate \( g_R \) and degrades at rate \( d_R \). This mRNA produces a protein \( A \) which acts as a repressor and controls the production rate of the mRNA. The production rate of \( A \) is thus proportional to \([R]\) and its degradation rate is \( d_A \). The intracellular dynamics is described by the rate equations

\[
\begin{align*}
\dot{[R]} &= \frac{g_R}{1 + k[A]} - d_R[R] \\
\dot{[A]} &= g_A[R] - d_A[A].
\end{align*}
\]  

(2)

where the dots represent time derivatives, namely \( \dot{[R]} = d[R]/dt \). These equations have two steady state solutions, however, only one of them is relevant because the other exhibits negative population sizes. The relevant solution is

\[
\begin{align*}
[R] &= \frac{d_A}{2g_A} \left[ \sqrt{\frac{1}{k^2} + \frac{4g_A g_R}{d_A d_R k^2}} - \frac{1}{k} \right] \\
[A] &= \frac{1}{2} \left[ \sqrt{\frac{1}{k^2} + \frac{4g_A g_R}{d_A d_R k^2}} - \frac{1}{k} \right]
\end{align*}
\]  

(3)

and the convergence to this solution is fast [Rosenfeld et al., 2002]. However, these equations do not take into account explicitly the chemical mechanism which enables the regulation. In this mechanism, one of the \( A \) proteins binds to the repression site on the DNA and inhibits the mRNA production. This protein should be subtracted from the population of free proteins in the cell, which Eq. (2) does not do. In addition, the constant \( k \) in the Hill function captures only the steady state repression rate and not its dynamical behavior.

The dynamics of the repression mechanism can be incorporated into the rate equation by taking the bound protein as a third component in the reaction network. This gives rise to three dynamic equations:

\[
\begin{align*}
\dot{[R]} &= g_R(1 - [r]) - d_R[R] \\
\dot{[A]} &= g_A[R] - d_A[A] - \alpha_0[A](1 - [r]) + \alpha_1[r] \\
\dot{[r]} &= \alpha_0[A](1 - [r]) - \alpha_1[r]
\end{align*}
\]  

(4)
where \([r]\) represents the average population of bound repressors in a cell. Since there is only a single repression site in each cell, \([r]\) is limited to the range \(0 \leq [r] \leq 1\). In fact, it represents the fraction of time in which the repressor site on the DNA is occupied by a bound repressor. The average productivity of the DNA in producing mRNA’s is proportional to \(1 - [r]\). The binding coefficient \(\alpha_0\) is the rate in which a free protein becomes bound. This rate should be multiplied by the number of free proteins and by the average number of unoccupied repression sites per cell, \(1 - [r]\). The desorption coefficient \(\alpha_1\) is the rate in which a bound protein leaves the repression site. The reduced rate equation set given by Eq. (2) is an approximation to the extended set of Eq. (4) in the following manner: when \(\alpha_0\) and \(\alpha_1\) are large compared to other rate constants, \([r]\) approaches steady state much faster than \([A]\) and \([R]\). In this case, it is justified to assume that \([r]\) is in quasi steady state and impose \([r] = 0\). This gives the steady state solution

\[
[r] = \frac{\alpha_0[A]}{\alpha_1 + \alpha_0[A]}.
\]  

(5)

Substituting this solution into Eq. (4) gives the reduced set of Eq. (2), with \(k = \alpha_0/\alpha_1\). This implies that Eq. (3) is the steady state solution of Eq. (4) as well. This solution is stable and there are no oscillations for any values of the parameters. However, the time dependent solutions of Eq. (2) and of Eq. (4) are not the same. Whereas Eq. (2) assumes rapid convergence of \([r]\) into its steady state, Eq. (4) holds also in case that the relaxation time is long. In Figs. 1 and 2 we compare the dynamics described by the two sets of equations. The rate constants are \(g_R = 0.05\), \(g_A = 0.06\), \(d_R = 0.02\), \(d_A = 0.02\), \(\alpha_0 = 0.001\) and \(\alpha_1 = 0.001\) (all in units of \(s^{-1}\)). These rates represent typical transcription and translation times, which are of the order of 10 to 20 seconds. Typical half-life times of proteins and mRNA’s vary in the range of several minutes (Elowitz and Leibler, 2000). All these time scales are much shorter than the cycle time, which is typically around 30 minutes.

The dynamical behavior of \([A]\) turns out to be different in the two sets of equations. The deviations from steady state are much larger in the extended set of equations. The dynamics is also highly dependent on the initial condition of \([r]\) which is an additional degree of freedom that does not exist in the reduced set. In Fig. 1 where the initial condition is \([r] = 0\), the extended set shows an over-shoot in \(A\) production, while in Fig. 2 where the initial condition is \([r] = 1\), it shows an under-shoot in \(A\) production.

In some cases the regulation of the production of a protein \(A\) is mediated by a more complex molecule. For example, the repressor may be a molecule \(D\) which is a dimer of \(A\) molecules produced by the reaction \(A + A \rightarrow D\). The standard way of modeling such a circuit is to modify the repression term (the Hill function) in Eq. (2) to
Fig. 1. Intracellular dynamics as calculated by Eq. (2) (solid line) and by Eq. (4) (dashed line). The average amount of bound proteins $[r]$ is also shown (dotted line). The initial conditions are $[A] = 3$ and $[r] = 0$.

\[
\dot{[R]} = \frac{g_R}{1 + k[A]^2} - d_R[R]
\]
\[
\dot{[A]} = g_A[R] - d_A[A] - 2\alpha_2[A]^2
\]
\[
\dot{[D]} = \alpha_2[A]^2 - d_D[D] - \alpha_0[D](1 - [r]) + \alpha_1[r]
\]

For this system the extended set includes equations for $[R]$ and $[A]$, as well as for the dimer (repressor) population $[D]$ and for the bound repressor $[r]$. The equations take the form:

\[
\dot{[R]} = g_R(1 - [r]) - d_R[R]
\]
\[
\dot{[A]} = g_A[R] - d_A[A] - 2\alpha_2[A]^2
\]
\[
\dot{[D]} = \alpha_2[A]^2 - d_D[D] - \alpha_0[D](1 - [r]) + \alpha_1[r]
\]
As in the ordinary case in which the repressor is the protein $A$ itself, the inhibition term $1 - [r]$ in Eq. (7b) is equal to the Hill function of Eq. (6) in the limit of rapid relaxation of $[r]$. In this case $k = \alpha_0 \alpha_2 / (\alpha_1 d_D)$. However, when the repressor is the dimer $D$, there is an additional term in Eq. (7c) which has no analogue in Eq. (6). This term gives rise to a difference in the results of the reduced and the extended sets even in the steady state solution, as shown in Fig 3. The steady state solution of the extended set is stable and exhibits no oscillations. The parameters used in Fig 3 are the same as in Figs. 1 and 2, and the additional parameters are the degradation rate of dimers, $d_D = 0.02 \text{ (s}^{-1}\text{)}$, and the production rate coefficient of dimers, $\alpha_2 = 0.01 \text{ (s}^{-1}\text{)}$. The latter coefficient is determined by the diffusion rate of proteins in the cell. A
related quantity, namely, the time it takes for a protein to diffuse across the cell was recently measured (Elowitz et al., 1999) and found to be of the order of one second. The inverse of this time can be used as an upper bound for the production rate coefficient $\alpha_2$.

The reduced set of equations does not take into account explicitly the dimer population, which is responsible for the repression. Both Eqs. (6) and (7) do not take into account the fact that one needs at least two $A$ proteins simultaneously in the cell in order to produce a dimer. Therefore, when the population of $A$ proteins goes down to order 1 both equations fail and the master equation formalism is required.
3 The Master Equation

Rate equations are used to describe the dynamics of the average number of entities (such as proteins) in large populations such as those handled in in vitro experiments. In these equations it is assumed that the densities of substances are continuous variables that behave in a deterministic fashion. This approach is not suitable for genetic regulatory networks when the populations of the relevant species in a single cell are small (Gillespie, 1977, Nicolis and Prigogine, 1977, Ko, 1991, 1992, McAdams and Arkin, 1999, Szallasi, 1999, Gibson and Mjolsnes, 2001). In this case one should take into account the discrete nature of the populations and the fact that for small populations the fluctuations become significant. In negative regulatory systems there is a population of free repressors in the cell. In addition, there is a single repression site on the DNA where a single repressor molecule may bind. Therefore, each repression site can be either occupied by a repressor molecule (where \( r = 1 \)) or vacant (\( r = 0 \)). Thus, \( r \) cannot take any intermediate values. In such cases fluctuations may have an important impact on the processes involved and their dynamics should be described in more detail.

One of the approaches suggested is the use of stochastic simulations which take into account the dynamics of all participating substances (Gillespie, 1977, McAdams and Arkin, 1997, Morton-Firth and Bray, 1998, Gibson and Bruck, 2000). The difficulty with these simulations is that they are based on the accumulation of large amounts of statistical data, and thus require extensive computer simulations. Thus, this approach is not always feasible in the case of complex networks which involve a large number of proteins. A complementary approach is based on direct integration of the master equation (McAdams and Arkin, 1997, 1999, Paulsson and Ehrenberg, 2000, Paulsson, 2001, Kepler and Elston, 2001, Paulsson, 2002, 2004). This approach takes into account the probability distribution of all possible states of the system, and not only the average values as in the rate equation approach. It captures the time evolution of the probabilities of all the microscopic states of the system.

We now apply the master equation approach to study the negative autoregulatory circuit of Eq. 4. We denote the number of copies of the free protein \( A \) by \( n_A \) and of the mRNA by \( n_R \). The number of proteins \( A \) which are bound to the repression site on the DNA is given by \( n_r \). For a single repression site \( n_r \) can only take the values 0 or 1. The master equation follows the time evolution of the probability distribution \( P(n_R, n_A, n_r) \). It takes the form

\[
\dot{P}(n_R, n_A, n_r = 1) = g_A n_R [P(n_R, n_A - 1, 1) - P(n_R, n_A, 1)]
\]
\begin{align*}
\dot{P}(n_R, n_A, n_r = 0) &= g_An_R[P(n_R, n_A - 1, 0) - P(n_R, n_A, 0)] \\
&+ d_R[(n_R + 1)P(n_R + 1, n_A, 1) - n_RP(n_R, n_A, 1)] \\
&+ d_A[(n_A + 1)P(n_R, n_A + 1, 1) - n_AP(n_R, n_A, 1)] \\
&+ \alpha_0((n_A + 1)P(n_R, n_A + 1, 0) \\
&- \alpha_1P(n_R, n_A, 1)) \\
\dot{P}(n_R, n_A, n_r = 1) &= g_R[P(n_R - 1, n_A, 0) - P(n_R, n_A, 0)],
\end{align*}
\tag{8b}

where the two cases of $n_r = 0$ and $n_r = 1$ are presented separately. The first terms in the equations describe the formation of a new protein. The second and third terms describe the degradation of the mRNA and the protein, respectively, while the fourth and fifth terms describe the binding and unbinding of a protein to the repression site on the DNA. Eq. (8b) also includes a term that corresponds to the formation of a new mRNA (not possible in the repressed case). These equations can be integrated numerically in order to obtain the time dependence of the probability distribution. It can also be solved for steady state by taking $\dot{P}(n_R, n_A, n_r) = 0$.

The master equation provides all the moments of the distribution $P(n_R, n_A, n_r)$ and their time dependence. For example, the average population of proteins $A$ is given by

\begin{equation}
\langle n_A \rangle = \sum_{n_R=0}^{n_R^{\text{max}}} \sum_{n_A=0}^{n_A^{\text{max}}} \sum_{n_r=0}^{1} n_A P(n_R, n_A, n_r)
\end{equation}

where $n_R^{\text{max}}$ and $n_A^{\text{max}}$ are the cutoff values that provide upper bounds on the populations of mRNA molecules and $A$ proteins in the cell, respectively. The repression site can be either occupied ($n_r = 1$) or unoccupied ($n_r = 0$).

Solving the master equation under steady state conditions for systems with different rate constants we calculated the appropriate averages, and compared the results with the rate equations.

In Fig. 4 the average levels of free proteins, mRNA molecules and bound protein (repressor) in the cell (at steady state), are shown vs. $\alpha_0$, as obtained from the master equation (solid line) and the rate equations (dashed line). The rate equations turn out to overestimate the average level of proteins and mRNA molecules, by a factor of 2-4 for systems with low copy number of proteins. On the other hand, when the average number of proteins in the cell is large, the results of the rate equations and master equation coincide.
Mathematically the discrepancy between the results of the rate equations and the master equations is due to non-linear terms such as the term that describe the attachment rate of proteins to the repression site. In the rate equation, this term is given by $\alpha_0[A](1-[r])$, namely as a product of averages (first moments). In the master equation it is given by the second moment $\alpha_0\langle n_A(1-n_R)\rangle$. The formation of dimers is also described by a nonlinear term. In the rate equations this term is given by $\alpha_2[A]^2$, namely it depends only on the first moment. In
the master equation it is given by $\alpha_2 \langle n_A^2 \rangle - \alpha_2 \langle n_A \rangle$, thus it depends on both the first and second moments.

The simple networks studied here can be considered as modules or motifs in complex genetic networks. However, the simulation of complex networks using the master equation is difficult. This is due to the proliferation in the number of equations as the number of components (mRNA’s and proteins) increases. Consider, for example, a network that involves three protein species, A, B and C. The master equation is written in terms of the probabilities $P(n_A, n_B, n_C)$ of having a certain population of proteins. The population size of each protein is limited by an upper cutoff. For example, the population of protein A takes the values $n_A = 0, 1, \ldots, n_{A}^{\text{max}}$. Clearly, the number of equations increases exponentially with the number of species, making this approach infeasible for complex networks.

However, typically these networks are sparse, namely most pairs of proteins do not interact with each other. This feature makes it possible to divide the master equation into several sets of equations, each set including only a small number of protein species. For example, if proteins B and C do not interact, the master equation described above can be broken into two sets that involve $P_{AB}(n_A, n_B)$ and $P_{AC}(n_A, n_C)$. In the case of large and sparse networks this dramatically reduces the number of equations and thus enables the simulation of complex networks using the master equation. This technique, named the multi-plane method, was recently proposed in the context of chemical reaction networks on interstellar dust grains [Lipshtat and Biham, 2004]. The mathematical structure of these networks is similar to that of genetic networks. Thus, the multi-plane method is perfectly applicable for the simulations of complex genetic networks. The similarity between the two systems is briefly discussed below.

### 4 Discussion: Genetic Networks and Grain-Surface Chemistry

Processes which exhibit a similar mathematical structure to the genetic network dynamics appear in the context of chemical reaction networks on interstellar dust grains. The chemistry of interstellar clouds consists of reactions taking place in the gas phase as well as on the surfaces of dust grains [Hartquist and Williams, 1995]. It turns out that the most abundant molecule in the Universe, namely molecular hydrogen does not form in the gas phase but on dust grain surfaces [Gould and Salpeter, 1963; Hollenbach and Salpeter, 1971; Hollenbach et al., 1971]. These grains are made of amorphous silicate and carbon compounds and are of sub-micron size. In addition to the formation of molecular hydrogen, these grains support complex reaction networks that produce a variety of molecules that consist of hydrogen, oxygen, car-
Table 1
Analogies between the processes of surface chemistry and of gene regulation.

| Description          | Surface chemistry                  | Gene regulation                  |
|----------------------|------------------------------------|----------------------------------|
| system               | dust grain                         | cell                             |
| break-up mechanism   | grain fragmentation                | cell division                    |
| mobility             | surface diffusion                  | diffusion in cell                |
| addition $∅ \rightarrow A$ | flux $F$                           | transcription $g_R$, production $g_A$ |
| removal $A \rightarrow ∅$ | desorption $W$                     | degradation $d_R, d_A$            |
| typical reaction     | $A + B \rightarrow C + D$          | $A \rightarrow A + B$            |
| feedback regulation  | rejection: $F(1 - \theta)$         | repression: $g_R(1 - [r])$       |

Consider a dust grain exposed to a flux of atomic and molecular species such as H, O, OH and CO. Atoms and molecules that hit and stick to the grain hop as random walkers between adsorption sites on its surface. When two atoms/molecules encounter one another they may react and form a more complex molecule. The rate equations that describe the reaction networks on grains include flux terms, desorption terms and reaction terms. The flux terms represent the flow of atoms and molecules from the gas phase onto the surface. The desorption rates are proportional to the population sizes of atoms and molecules on the grains, while the reaction terms are proportional to the products of the population sizes of the reactive species. In general, the rate equations resemble those that describe genetic networks. The analogy between the two systems is summarized in Table 1. In both systems reactive species are added, diffuse, react and removed. The system itself may break up (cell division or grain fragmentation), dividing the population of reactive species into two sub-populations. Both systems exhibit some kind of negative feedback. In genetic networks this is provided by the repression circuit, in which the rate of attachment of proteins to the repression site is given by $\alpha_0[A](1 - [r])$. Certain surface reaction systems exhibit the Langmuir rejection behavior, in which atoms from the gas phase that hit the surface in the vicinity of an already adsorbed atoms are rejected. The flux term $F$ is then modified to the form $F(1 - \theta)$, where $\theta$ is the coverage, namely the fraction of adsorption sites on the surface that are occupied by adsorbed atoms. In the context of grain-surface chemistry, low copy numbers are obtained in the limit of small grains under conditions of low flux. In this limit the master equation is required (Biham et al., 2001, Green et al., 2001, Biham and Lipshtat, 2001).
For complex reaction networks of multiple species, the master equation becomes infeasible due to the proliferation in the number of equations. In this case, the multi-plane method is used in order to keep the number of equations at a tractable level [Lipshtat and Biham 2004].

5 Summary

We have considered the rate equation and master equation approaches to the modeling of genetic networks. In particular, we have studied the temporal evolution of the population of mRNA and proteins in simple negative autoregulated genetic networks. As long as the populations of all the reactive components of the network are not too small, rate equations provide a good quantitative description of the network dynamics. However, once the populations of the mRNA or proteins are reduced to order 1 or less, rate equations are no longer suitable and the master equation is needed. This is due to the fact that the rate equations involve only average quantities, while the master equation takes into account the discrete nature of the populations as well as the fluctuations. The simple networks studied here can be considered as modules or motifs in complex genetic networks. The simulation of complex networks using the master equation is difficult, because the number of equations quickly proliferates. The multi-plane methodology, recently developed in the context of grain-surface chemistry, that tackles this problem is briefly described. Finally, the analogy between genetic networks and grain-surface chemistry is discussed.

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