Multi-scale analysis on soil improved by alkali activated binders

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Abstract. In the present paper, the use of alkali activated binders to improve engineering properties of clayey soils is presented as an alternative to traditional binders such as lime or cement. An alkali-activated fly ash and its chemo-physical evolution has been monitored at increasing curing times by means of X-Ray Diffraction and Scanning Electron Microscopy. Alkali-activated binder has been mixed with soil for evaluating the improvement of its mechanical behaviour. One-dimensional compression tests on treated samples have been performed with particular reference to effects induced by binder content and curing time. Test results showed a high initial reactivity of the alkali activated systems promoting formation of new mineralogical phases responsible of the mechanical improvement of the treated soil.

1 Introduction

The development and the use of environmentally friendly binders as an alternative solution to traditional binders such as lime or cement in a low carbon agenda is of prime importance particularly in the construction sector. The use of novel and efficient binders for geotechnical applications is a promising issue in terms of sustainability since it reduces the carbon footprint and allows reusing secondary by-products such as artificial pozzolans. These by-products can be involved in soil improvement as cementing agents if properly activated, inducing a mechanical improvement of natural soils, not suitable for construction purposes.

Alkali activated binders represent a viable, sustainable alternative to the use of ordinary stabilizing agents for soil improvement [1]. Differently from the use of lime and cement for soil improvement [2-5], experimental research about the use of alkali activated binders in soil improvement is still limited; nevertheless, recent studies highlight the relevant potential of novel binders for geotechnical purposes. Cristelo et al. [6] researched the optimum fly ash - based alkali activated binder for the improvement of soil to be used in rammed earth construction through a parametric analysis using laboratory tests. Rios et al. [7] compared the mechanical behaviour of samples of sand improved by an alkali activated fly ash binder and the similar samples improved by cement, highlighting the effectiveness of the alkali activated binders in increasing the shear strength properties of the soil.

Alkaline activation of by-products is the consequence of a chemical reaction between an aluminosilicate source with an alkaline solution (i.e. sodium hydroxide, sodium silicate). The aluminosilicate source is formed by precursor materials like natural pozzolan (e.g., pyroclastic soils), or artificial pozzolans (e.g., fly ash, silica fume, steel sludge). The chemo-physical evolution occurring after the activation promotes the formation of cementitious compounds [8-9]. The reaction mechanism is promoted by the alkaline solution, which enables the dissolution of the aluminosilicate source (precursor) and the subsequent precipitation of gel phases, which condense in a three-dimensional aluminosilicate network [10-12].

In the present study, an insight into the alkaline activation of an artificial pozzolan (fluidal bed combustion fly ash) and its effects on the improvement of the mechanical behaviour of a clayey soil have been reported. A multi scale analysis have been performed to highlight the link between the ongoing reactions and the macroscopic evolution of soil properties. The chemo-physical evolution of the binder has been monitored at increasing curing times by means of X-Ray Diffraction analysis and Scanning Electron Microscopy (SEM). The macroscopic behaviour of the alkali-activated binder treated soil has been investigated by means of one-dimensional compression tests. The effects of binder content and curing time on the mechanical performance of treated samples have been considered.

2 Material and experimental procedures

2.1. Material

The artificial pozzolan used in the present work is a fluidal bed combustion fly ash supplied by a power plant located in Italy. Raw material is composed by vitreous and crystalline phases, which include calcium-containing minerals (anhydrite, calcite) and other minerals (quartz,
The chemical composition of fly ash is reported in Table 1. The sodium silicate solution was supplied by Woellner (Germany) with a SiO₂/Na₂O mass ratio of 1.7. Speswhite kaolin from deposits in the South West of England was the soil considered for the experimental investigation. The specific gravity Gₛ is 2.6 and its surface area, determined by BET, is 14 m²/g. The pH value is 4.6 and plastic limit and liquid limit are respectively 32% and 70%, with a plasticity index IP equal to 38%. X-Ray diffraction pattern is shown in Figure 1. The soil is mainly composed of kaolinite clay minerals with small amounts of quartz and muscovite. The chemical composition of the soil is given in Table 2.

Table 1. Chemical composition of fly ash.

| Constituent | Percentage (%) |
|-------------|----------------|
| SiO₂        | 19.80          |
| Al₂O₃       | 39.4           |
| CaO         | 5.2            |
| K₂O         | 1.8            |
| Fe₂O₃       | 7.14           |

Table 2. Chemical composition of Speswhite kaolin.

| Constituent | Percentage (%) |
|-------------|----------------|
| SiO₂        | 53.80          |
| Al₂O₃       | 43.75          |
| CaO         | 0.02           |
| K₂O         | 1.45           |
| TiO₂        | 0.05           |

Fig. 1. X Ray diffraction pattern of Speswhite kaolin.

2.2 Samples preparation

Alkali activated binder (FA100% in the following) was prepared by mixing fly ash, alkaline solution and water in fixed proportions. Alkaline solution/fly ash mass ratio was kept constant and equal to 0.5. Additional deionised water was provided to the system in order to guarantee an effective mixing and a sufficient liquid phase amount for the dissolution of aluminosilicate source. The deionised water/solid ratio, by mass, was selected equal to 0.5. Soil samples treated with alkali-activated binder were prepared by considering increasing fly ash percentages between 25° and 35°, corresponding to new poorly crystalline compounds resulting from alkali-activation reactions. A broad reflection between 25° and 35°, corresponding to new poorly crystalline compounds resulting from alkali-activation reactions, was observed in alkali activated fly ash after 60 days of curing. SEM observations on raw and alkali activated fly ash after 60 days of curing are shown in Figures 3 and 4. The raw sample is characterised by particles of different shape and size. The vitreous phase is made of some bigger grains (Figure 3a; 3b-1) and aggregate of small spherical particles (Figure 3b-2; 3c). A coating of gel hydrates

2.3 X-Ray diffraction analysis

Mineralogical composition of samples was investigated by X-Ray analysis performed on randomly oriented powder using a Brucker AXS D8 Advance Diffractometer with CuKα (λ = 0.154 nm) radiation and a step size of 0.017°. Samples were dehydrated before testing by freeze-drying technique [13].

2.4 Scanning Electron Microscopy (SEM)

Surface state modifications of samples due to alkaline activation process have been examined through Scanning Electron Microscopy (SEM) by using Hitachi SU5000 microscope. Raw and treated samples have been dehydrated by freeze-drying technique. A pre-treatment gold coating has been applied for SEM observations.

2.5 One-dimensional compression tests

One-dimensional compression tests have been performed on alkali activated samples cured for 24 hours, 7 and 28 days. Tests have been performed in standard oedometer cells, where vertical stress was conventionally applied in successive steps (Δσ/σ = 1) within the stress interval 10 000 kPa. Micrometer dial gauges with an accuracy of 0.001 mm have been used to measure vertical displacements.

3 Results

X-Ray diffraction patterns of raw and alkali activated fly ash binder (FA100%) are shown in Figure 2. Mineralogical composition of fly ashes is modified by alkaline activation process. Consumption of crystalline phases such as anhydrite (CaSO₄) and aluminophases (MgAl₂O₄) is evidenced by disappearance of their characteristic peaks after 60 days of curing. Precipitation of new mineralogical phases such as thenardite (Na₂SO₄) seems to be consistent with the release of sulphate from anhydrite dissolution and its subsequent reaction with sodium provided by silicate solution. A broad reflection between 25° and 35°, corresponding to new poorly crystalline compounds resulting from alkali-activation reactions, was observed in alkali activated fly ash after 60 days of curing. SEM observations on raw and alkali activated fly ash after 60 days of curing are shown in Figures 3 and 4. The raw sample is characterised by particles of different shape and size. The vitreous phase is made of some bigger grains (Figure 3a; 3b-2; 3c), spherical particles (Figure 3b-1) and aggregate of small particles (Figure 3b-3). A coating of gel hydrates...
Fig. 2. X Ray diffraction patterns of raw fly ash and alkali activated fly ash (FA100%) at 60 days of curing time.

Fig. 3. SEM observations of raw fly ash.
Fig. 4. SEM observations of alkali activated fly ash at 60 days of curing.

on particles surface is observed at 60 days of curing (Figure 4a, b). After the treatment, surface state of the sample is clearly modified. The alteration of glassy/vitreous phases induced by the alkaline environment is clearly evidenced by SEM on raw and alkali activated samples at higher magnifications (Figure 3d and 4d).

1D compressibility curves of alkali activated binder treated soil (KFA40%) at increasing curing times, namely 24 h, 28 and 60 days are reported in Figure 5. Addition of alkali activated binder induces an overall reduction of compressibility of treated samples, with reduced volume strains for reference vertical stresses. The reduction is more relevant for increasing curing times, showing the stiffer behaviour of treated samples although the behaviour of treated samples does not evolve with time after 28 days of curing. Coupled with reduction of compressibility, all the samples show an increase of yield stress. Transition from a reversible behaviour to an irreversible one is shifted (yield stress) to higher vertical stresses, inducing an advantageous behaviour of the improved material for construction purposes. In Figures 6, compressibility curves of treated samples are reported as function of binder contents. KFA samples prepared at 20% and 40% of alkali activated fly ash binder (KFA20% vs. KFA40%) and cured for 24 hours before testing (Figure 6a) show no relevant changes in the compressibility curves. The improvement of mechanical response of treated samples after 28 days of curing is more relevant when increasing the amount of binder; both reduction of compressibility and yield stress increase depend on binder content. Differently from the shorter curing time, the post-yield behaviour shows an increased slope of the curve, highlighting a more evident de-structuration of the samples at increasing vertical stresses. A comparison between mechanical performance induced by alkali activated binder (KFA40%) and ordinary Portland cement (KC40%) has been shown in Figure 7 for samples prepared with the same binder content and cured for 60 days, in order to highlight the role of type of binder on the mechanical improvement of treated soil. Results show a higher compressibility reduction and yield stress increase for KFA treated samples compared to cement treated samples (KC40%). For stress levels higher than yield stress, cement treated samples show a higher compressibility coefficient (i.e. slope of the compressibility curve) with respect to alkali-activated samples, depending on the de-structuration stage induced by load increase. Stress levels achieved during the tests were not high enough to induce the complete de-structuration of both treated samples.

Fig. 5. One-dimensional compression tests on untreated and KFA40% treated kaolin as function of curing time.
Fig. 6. One-dimensional compression tests on treated samples as function of binder content - KFA20% vs. KFA40%: a) 24h of curing; b) 28 days of curing.

Fig. 7. One-dimensional compression tests on treated samples as a function of type of binder (KFA40% vs. KC40%) at 60 days of curing.

4 Conclusions

An insight into the mechanical improvement induced by an alkali-activated binder based on the activation of a fly ash on a clayey soil has been presented. The experimental multiscale investigation on the chemo-physical evolution of the binder show the high reactivity of the alkali activated fly ash to dissolve the aluminosilicate source and to promote the precipitation of new mineralogical phases with cementitious properties. Experimental evidences at microscale have been linked to the macroscopic behaviour of a treated soil. One-dimensional compression tests performed on treated sample highlight the effectiveness of alkali activated binder to promote an improvement of the mechanical behaviour of treated soil. An increase of the compressive strength and of the yield stress soil was observed since the very short term, whose extent depends on curing time and binder contents. The efficiency of treatment has been highlighted by comparing the mechanical performance induced by alkali-activated binders with the one promoted by ordinary Portland cement. A marked improvement of the mechanical behaviour of soil is induced by alkali-activated binder, representing a viable, sustainable alternative to the use of ordinary stabilizing agent for soil improvement.

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