Astrophyllite–alkali amphibole rhyolite, an evidence of early Permian A-type alkaline volcanism in the western Mongolian Altai

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A dyke of alkali rhyolite intrudes the Tsetseg and Zuun Nuruu volcanosedimentary sequence of Ordovician–Silurian age (Hovd Zone, Central Asian Orogenic Belt) at the Botgon bag, Mankhan Soum, Hovd District in Western Mongolia. The rock consists of quartz and K-feldspar phenocrysts set in fine-grained groundmass composed of quartz, K-feldspar, albite, blue alkali amphibole (riebeckite–arfvedsonite containing up to 1.94 wt. % ZrO2), tiny brown radial astrophyllite, annite and accessory zircon, ilmenite, fluorite, monazite, hematite, chevkinite and bastnäsite. Astrophyllite has unusual, highly ferroan composition and occurs as two sharply bound zones of astrophyllite I and II with the average empirical formulae: 

(K1.71 Na0.01Rb0.08Cs0.01) (Na0.93Ca0.07) (Fe2+6.52Mn0.31Zn0.06) (Ti0.84Zr0.50Nb0.55) Si7.68Al0.32 O26 (OH)3.78 F0.66 (astrophyllite I, Zr–Nb-rich); (K1.52Rb0.07) (Na0.81Ca0.19) (Fe2+6.31 Mn0.28Zn0.06) (Ti1.28Nb0.30Zr0.28) Si7.68Al0.32 O26 (OH)2.85 F0.67 (astrophyllite II).

Geochemically, the rhyolite corresponds to strongly fractionated silicic alkaline A-type (ferroan) magmatic rock with 75.5–75.9 wt. % SiO2, 4.4 wt. % K2O, 3.9–4.3 wt. % Na2O and 1.98–2.23 wt. % Fe2O3, poor in CaO (0.26–0.37 wt. %), MgO (0.01–0.11 wt. %), and P2O5 (0.01 wt. %). The rock is enriched in Zr, Nb, Ta, Sn, Y, Rb, Cs, and exhibits a pronounced negative Eu anomaly (Eu/Eu* = 0.03–0.05). The conventional whole-rock K–Ar geochronology yielded an age of 299.9 ± 9.1 Ma (1σ), which indicates latest Carboniferous or early Permian extension associated with the A-type alkaline volcanic activity.

Keywords: rhyolite, A-type, astrophyllite, K–Ar geochronology, CAOB

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1. Introduction

The Central Asian Orogenic Belt (CAOB, Şengör et al. 1993; Jahn et al. 2000) is located between Uralian Orogen in the west, Siberian Craton in the north, Tarim Craton in the southwest and Sino-Korean Craton in the south. The CAOB represents the largest accretionary system on the Earth (~8000 × 6000 km) formed as a result of the subduction and closure of the Paleo-Asian Ocean (Zonenshain et al. 1990; Şengör et al. 1993; Jahn et al. 2000; Buslov et al. 2001). The CAOB is a complex “collage” of continental blocks, accretionary wedges, magmatic arcs, back-arc and fragments of oceanic crust amalgamated during the Paleozoic to Mesozoic eras (Şengör et al. 1993).

The Late Paleozoic rift-related intrusive and extrusive acid igneous rocks are relatively common in northern China and south-western Mongolia (Wang ZG et al. 1993; Litvinovsky et al. 2002; Wu et al. 2002; Chen and Jahn 2004; Jahn 2004; Chen and Arakawa 2005; Wang Q et al. 2007; Burianek et al. 2012; Liu et al. 2013; Mao et al. 2014). These generally undeformed, anorogenic-type intrusions provide evidence of the final amalgamation of terranes in this part of the CAOB (Jahn 2004; Liu et al. 2013; Mao et al. 2014). This paper is intended to provide new geochemical and geochronological data on the astrophyllite- and alkali amphibole-bearing rhyolite dyke cutting the Lower Paleozoic volcanosedimentary and plutonic sequence in the Hovd Zone in Western Mongolia.

2. Geological setting

The study area is situated near the boundary between two terranes: (1) the Lake Zone to the east and (2) the Hovd Zone to the west (Badarch et al. 2002, Fig. 1). These domains are separated from each other by an important crustal-scale boundary, which was interpreted as a wide strike-slip fault zone (Tomurtogoo 1997).

The Lake Zone is a composite terrane, which contains Neoproterozoic to Cambrian volcanosedimentary complexes and ophiolites intruded by Late Cambrian–Middle Ordovician syn- to post-tectonic gabbro, diorite, granodiorite and granite (Kovalenko et al. 1996; Kröner et al. 2010; Soejono et al. this volume). The Hovd Zone is interpreted as a part of the accretionary wedge (e.g., Xiao et al. 2004, details in Soejono et al. this volume) and is composed of several lithostratigraphic units of documented or supposed Ordovician to Silurian ages.
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(Baatarhuyag and Gansukh 1999). Also, the existing geochronological studies mostly from the Chinese Altai suggest these rocks to have been deposited during the Ordovician to the Silurian (Sun et al. 2008; Long et al. 2009). Lithologically, the Hovd Zone is composed mainly of pelitic schists intercalated with basic volcanic rocks and fossiliferous carbonate lenses. Greenschist- to amphibolite-facies metamorphism occurred during an Early Devonian event as shown by a study of zircon in the Habahe Group in the adjacent Tseel Terrane (Jiang et al. 2010, 2012). This sequence is crosscut by various granitic, dioritic to gabbroic intrusions, mostly syn-metamorphic. Numerous U–Pb zircon ages from similar calc-alkaline granitic rocks of the Habahe Group suggest Devonian to early Carboniferous age of the major magmatic pulse (Jiang et al. 2010). The youngest formation consists of Lower Carboniferous siliciclastic sediments deposited in a syn-orogenic basin.

3. Methods

3.1. Field work

Field work was carried out and relevant material and samples collected in June–July 2014, and 2015. During the field mapping, all important structural data were documented, and the KT-10 kappameter was used to measure the magnetic susceptibility.

3.2. Whole-rock geochemistry

The whole-rock analyses were performed in Activation Laboratories (Actlabs; Toronto, Canada) using the procedure 4Lithoresearch (http://www.actlabs.com). Major-element concentrations were obtained by inductively-coupled plasma optical emission spectrometry (ICP-OES), trace-element concentrations were established by inductively-coupled plasma mass spectrometry (ICP-MS). Subsequently, the sample GZ195 was analyzed in Laboratories of the Czech Geological Survey in Prague (using the material pulverized in the Actlabs) to determine FeO, Fe2O3, Li2O, and F (Tab. 1). Geochemical data were recalculated and plotted by the GCDkit software package (Janoušek et al. 2006).

3.3. Electron microprobe analyses (EMPA)

Electron-microprobe analyses of minerals (sample GZ195) were carried out using a Cameca SX-100 electron microprobe in the Joint laboratory of the Masaryk University and of the Czech Geological Survey in Brno. The astrophyllite and amphibole were
measured at conditions: accelerating voltage 15 kV, beam current 10 nA and beam diameter 8 µm. The following lines, monochromators and standards were used: F (Kα, PC1, topaz); Si, K, Al (Kα, TAP, sanidine); Mg (Kα, TAP, pyrope); Na (Kα, TAP, albite); Rb (Lα, TAP, synthetic Rb-leucite); FeO (Lα, TAP, zircon); Ti (Kα, PET, anatase); Ca (Kα, PET, wollastonite); Cl (Kα, PET, vanadinite); Nb (Lα, PET, columbite-Fe); Cs (Lα, PET, pollucite); Fe (Kα, LIF, almandine); Mn (Kα, LIF, spessartine); V (Kα, LIF, ScVO4); Zn (Kα, LIF, gahnite). Chromium, P, Y, La, Dy and Yb were sought but not detected. Peak-counting times were 10 and 20 s for major elements and 30 and 40 s for minor to trace elements. Raw intensities were processed by X-PHI matrix correction (Merlet 1994). Natural minerals and stable synthetic compounds were used for analyses of other minerals; beam diameter was 3 µm for zircon and 5 µm for other minerals.

3.4. K-Ar geochronology

The age determination was made by conventional K–Ar method in ATOMKI, Debrecen, Hungary. Sample of rhyolite for the dating (GZ195) was prepared in the laboratories of the Czech Geological Survey in Prague. After the crushing, the whole-rock fraction of 0.125–0.30 mm (50–100 g) was washed several times in distilled water, and subsequently dried at 40°C.

3.4.1. Potassium determination

Approximately 0.05 g of finely pulverized sample was digested in acids (HF, HNO₃ and H₂SO₄) in Teflon beakers and finally dissolved in 0.2 M HCl. Potassium was determined by flame photometry with a Na buffer and Li internal standard using Industrial M420 type flame photometer. Multiple runs of inter-laboratory standards (Asia1/95, LP-6, HD-B1, GL-0) indicated the accuracy and reproducibility of this method to be within 2%.

3.4.2. Argon measurements

Approximately 0.5 g of the sample was wrapped in an aluminium foil and copper sieve preheated for about 24 h at 150–180°C in a vacuum. Argon was extracted under ultra-high vacuum conditions by RF (high frequency) induction heating and fusion of rock samples in Mo crucibles. The gas was purified by Ti sponge and SAES St 707 type getters to remove chemically active gas contaminants in addition to liquid nitrogen in cold trap to remove condensable gases. The extraction line is linked directly to a mass spectrometer (90° magnetic sector, 155 mm radius, equipped with a Faraday cap), used in static mode.

Argon isotope ratios were measured by the 38Ar isotope dilution mass spectrometric method, previously calibrated with atmospheric argon and international rock standards.

Experimental details and results of calibration have been described by Balogh (1985). The age of the samples is calculated using the decay constants suggested by Steiger and Jäger (1977). Analytical error is given at 68% confidence level (1σ) using the equation of Cox and Dalrymple (1967).

4. Results

4.1. Field observations

The rhyolite was sampled at two sites, both situated on the southern slope of a valley at the Botgon bag in Mankhan Soum, Hovd District in Western Mongo-
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Samples GZ195 and GZ330 were taken from two individual WNW–ESE striking ~2–5 m thick dykes, which cut with evident discordancy the gabbro, biotite schists, quartzite schists, quartzites, and porphyritic biotite granite (Fig. 2a). The schists, quartzite and gabbro belong to the Tsetseg and Zuun Nuruu formations of Silurian to Ordovician ages (Baatarhuyag and Gansukh 1999); the granite is likely of Devonian age. The studied rhyolite dykes are accompanied by a dyke swarm 300 m wide consisting of doleritic dykes up to ~15 m thick, all of which show the same strike (Fig. 1). However, while the dolerite dykes are common and distributed over much of the area mapped, alkali rhyolite was observed only at the locality studied.

4.2. Petrography

The rhyolite is fresh, massive, greyish in colour, and contains quartz and feldspar phenocrysts visible to the eye, 1–3 mm long (Fig. 2). The rock displays low magnetic susceptibility of 0.03–0.20 × 10⁻³ SI. In thin sections, a weak magmatic foliation is apparent (Fig. 3a). The mineral assemblage consists of quartz + K-feldspar + albite with minor alkali amphibole (2–3 vol. %), astrophyllite (1–2 %), anhitte (~1 %) and several accessory minerals.

Under the microscope, the rock consists of perthitic microcline phenocrysts 1–3 mm long (~3 vol. %), and minute quartz phenocrysts nearly 1 mm long (~ 5 vol. %). The groundmass (recrystallized glass, ~80 vol. %) is composed of a granophyric intergrowth of alkali feldspar and quartz, 0.2–0.3 mm in size, along with minute grains of quartz (Fig 3b). Strongly pleochroic blue alkali amphibole forms 0.01–0.2 mm long prismatic to stubby aggregates (Fig. 3c). Bright brown astrophyllite clusters into radial aggregates composed of prismatic to acicular crystals up to 0.3 mm long (Fig. 3d). Greenish-black flakes of anhitte are prismatic to irregular, up to 0.5 mm long. Accessory minerals include abundant zircon, minor monazite and REE-rich fluorite as well as rare Ca, Fe, Ti, Ce silicate, presumably chevkinite, and Ca-poor, F-rich, REE-bearing carbonate, possibly bastnäsite. Zircon forms euhedral stubby crystals or their angular fragments up to ~100 μm long. Monazite occurs as subhedral grains up to ~60 μm, and ?chevkinite was found as single 10 μm long inclusion in zircon. Inclusions of ?bastnäsite, ~1–5 μm across, and up to 25 μm long inclusions of REE-rich fluorite in alkali amphibole were identified by EDX. Hematite was encountered in a unique miarolitic structure filled with quartz, anhitte and minor hematite, the latter forming ~40 μm hexagonal tabular crystals.

4.3. Mineral chemistry

Sodic amphibole shows weak irregular zoning apparent in BSE images (Fig. 3e); the zoning has a character of metasomatic replacement. The dominant darker and most probably older amphibole has low Zr content (0.00–0.02 apfu), whereas he lighter zones correspond to younger, Zr-enriched compositions (0.05–0.22 apfu Zr). However, some intermediate compositions can be observed (Fig. 3e, Tab. 2). The sodic amphibole is Fe-dominant with low Mn(0.06–0.08 apfu), Mg (0.00–0.02 apfu), and Ti (0.02–0.12 apfu), see Tab. 2. The standard normalization procedures of EMPA data yielded non-stoichiometric formulae with Si = 8.02–8.22 apfu, perhaps reflecting the possible presence of Li in the amphibole. Direct measurement of Li by LA ICP-MS was not possible due to small size of the amphiboles. The tentative formula calculated, assuming Si = 8 apfu, yielded a minimum Li,O content 0.03–0.6 wt. %, which corresponds to 0.02–0.42 Li apfu (Tab. 2). The nomenclature of this alkali amphibole is therefore complicated due to variable (Na + K) occupancies (0.27–0.66 apfu), and possible presence of Li. Principally, the amphibole composition varies between...
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Fig. 3 Microscopic (a–d) and BSE images (e–f) of sample GZ195. a – Weak magmatic foliation (highlighted by grey lines) indicated by alternation of finer and coarser zones. The tiny semi-opaque inclusions are alkali amphibole, astrophyllite and annite (plane polarized light). b – Micrographic (granophyric) structure due to rapid cooling of the rhyolite melt (crossed nicols). c – A group of blue prismatic to stubby crystals of alkali amphibole associated with minor brownish prismatic to acicular crystals of astrophyllite (plane polarized light). d – Radial aggregates of acicular astrophyllite crystals associated with minor alkali amphibole. The associated minerals form intimate intergrowths of quartz and alkali feldspars (plane polarized light). e – Large corroded phenocryst of alkali amphibole with EMPA spots labelled by ZrO$_2$ concentrations (wt. %). The crystal of alkali amphibole displays zoning indicated by alternation of lighter streaks and patches, and relict darker domains. Lighter zones are strongly enriched in Zr (up to 1.48 wt. % ZrO$_2$), whereas dark domains contain only ~0.20 wt. % ZrO$_2$. Tiny white inclusions consist of REE-carbonate, possibly bastnäsite. f – Radial aggregate of astrophyllite with EMPA spots showing concentrations of ZrO$_2$ and Nb$_2$O$_5$ (in italic) (both in wt. %). Associated minerals include alkali amphibole, in places replaced by astrophyllite, quartz, potassium feldspar, and accessory ilmenite. The astrophyllite shows sector zoning, the darker zones represent mostly marginal parts of the crystals and have lower concentrations of both Nb and Zr. Abbreviations used: Hbl – alkali amphibole, Ast – astrophyllite, Qz – quartz, Kf – potassium feldspar, Fs – alkali feldspars, Ilm – ilmenite, Zrn – zircon.
Astrophyllite (both I and II) exhibits an atypical chemical composition. It is classified as astrophyllite on the basis of prevailing cation at C-site (Fe), but with high Fe\text{tot}/(Fe\text{tot} + Mn) ratio of 0.95–0.96 (Fig. 4a). The contents of Nb and Zr are also elevated, so that in the diagram depicting the astrophyllite group based on D-site occupancy, the compositions of astrophyllite I are plotted in the proximity of “Fe-zircophyllite” and niobophyllite fields (Fig. 4b) (Kapustin 1972; Piilonen et al. 2000, 2003; Sokolova 2012, see also Tab. 3).

The resulting average empirical formula of astrophyllite I (Zr–Nb-rich) corresponds to: (K\textsubscript{1.71} Na\textsubscript{0.01} Rb\textsubscript{0.08} Cs\textsubscript{0.01}) (Na\textsubscript{0.93} Ca\textsubscript{0.07}) (Fe\textsuperscript{2+} \texttextsubscript{0.52} Mn\textsubscript{0.31} Zn\textsubscript{0.05} Nb\textsubscript{0.02} Ti\textsubscript{0.84} Zr\textsubscript{0.50} Nb\textsubscript{0.02} Si\textsubscript{7.68} Al\textsubscript{0.32} O\textsubscript{26} (OH)\textsubscript{1.16} F\textsubscript{0.67}. In comparison, the average formula of astrophyllite II is: (K\textsubscript{1.21} Rb\textsubscript{0.07}) (Na\textsubscript{0.81} Ca\textsubscript{0.19}) (Fe\textsuperscript{2+} \texttextsubscript{0.52} Mn\textsubscript{0.31} Zn\textsubscript{0.05} Nb\textsubscript{0.02} Ti\textsubscript{1.28} Nb\textsubscript{0.50} Zr\textsubscript{0.28} Si\textsubscript{7.68} Al\textsubscript{0.32} O\textsubscript{26} (OH)\textsubscript{1.30} F\textsubscript{0.67}.

Anellite has a composition close to the end-member formula, slightly enriched in Rb and Zn: (K\textsubscript{1.01} Na\textsubscript{0.02} Rb\textsubscript{0.02}) (Fe\textsuperscript{2+} \texttextsubscript{2.12} Mg\textsubscript{0.03} Mn\textsubscript{0.03} Zn\textsubscript{0.04} Al\textsubscript{1.06} Si\textsubscript{4.14} OH\textsubscript{1.16} F\textsubscript{0.65}). The mica is purely ferroan, Mg(Mg\textsuperscript{2+} + Fe\textsuperscript{2+}) = 0.16. The phenocrysts of potassium feldspar (Or\textsubscript{98}Ab\textsubscript{2}) contain ~0.1 wt. % FeO, whereas the groundmass K-feldspar (Or\textsubscript{98}Ab\textsubscript{2}) is slightly richer in this oxide (0.31–0.39 wt. % FeO).

Plagioclase is represented by albite (Ab\textsubscript{95}Or\textsubscript{5}) slightly enriched in FeO (0.39 %), and RbO (0.11 wt. %). The concentrations of oxides of Ba, Sr, Cs, and P are below 0.02 wt. % in all feldspars.

Zircon contains elevated Th (0.012–0.035 apfu), U (0.003–0.007 apfu), Y (0.011–0.034 apfu), moderate Hf (0.013–0.014 apfu) and \Sigma REE (0.018–0.032 apfu).

### 4.4. Whole-rock geochemistry

The rhyolite (SiO\textsubscript{2} = 75.52 and 75.93 wt. %) is ferroan (FeO/(FeO + MgO) = 0.94–1.00 by weight; Fig. 5a), alkali-calcic (Na\textsubscript{2}O + K\textsubscript{2}O – CaO = 8.06–8.52 wt. %; Fig. 5b), and metaluminous (A/\Sigma CNK = 0.81–0.97), with...
a low MgO content (0.01–0.11 wt. %). The rhyolite is enriched in most of the LILE (e. g. Rb, K), HFSE (e. g. Th, Zr, Nb, Ta) and depleted in Ba, P, Sr and Ti when observed on the primitive mantle-normalized multi-element pattern (McDonough and Sun 1995; Fig. 6a). The chondrite-normalized REE pattern (Fig. 6b) shows higher proportion of LREE over HREE (LaN/YbN = 2.40–2.52), pronounced negative Eu anomaly (Eu/Eu* = 0.03–0.05) and a M-type tetrad effect (TE1–3 = 1.13–1.14; the degree of the lanthanide tetrad effect sensu Iber 1999).

The relatively high content of Zr (375–422 ppm), Nb (53–82 ppm), Ce (90–94 ppm; Fig. 7a), and Y (93–109 ppm; Fig. 7b), as well as the elevated 10000×Ga/Al ratios of 5.1 and 5.8 indicate an A-type affinity for the rhyolite (Whalen 1987; Bonin 2008).

In the diagram (Na₂O + K₂O – Fe₂O₃×5 – (CaO + MgO) × 5 (Fig. 7c) for classification of the A-type (ferroan) granites (Grebennikov 2014), the rhyolite falls along the boundary between the field of A₁-type granite (silicic rocks of within-plate geodynamic settings: ocean islands and continental rifts) and A₂-type granite (silicic rocks of post-collisional and late-collisional geodynamic settings as well as those generated at the late stages of evolution of hot rift structures).

Given the relatively low Y/Nb ratios (1.1–2.1), the studied rhyolites plot along, or below, the boundary between post-orogenic granites derived by partial melting of mafic lower continental crust or arc-type sources (A₂-type) and granites related to sources analogous to those of the ocean island basalts, further modified by crystal fractionation (A₁-type; Eby 1992; Fig. 7d). All these geochemical signatures correspond with other Permian acid igneous rocks generated in post-orogenic extensional tectonic environments from the southern Mongolia (e.g. Zhang et al. 2015, Blight et al. 2010a, b).
4.5. K–Ar dating

The studied sample (ryolite GZ195) yielded an age of 299.9 ± 9.1 Ma (1σ), which falls close to the Carboniferous/Permian boundary (see Tab 4). Since this rhyolitic dyke does not exhibit any significant alteration, and forms a small intrusive body, the established K–Ar age should be considered the emplacement age.

5. Discussion and conclusions

The results of detailed study show that the rhyolite has an anomalous mineralogy with regard to the occurrence of Zr-rich alkali amphibole (riebeckite–arfvedsonite, possibly ferroleakeite) and astrophyllite. Astrophyllite has extremely high Fe/(Fe+Mn) ratio of 0.95–0.96 (Fig. 4), not yet reported in minerals of the astrophyllite group (cf. Piilonen et al. 2000, 2003).

The geological position and chemical composition of the studied rhyolite dyke (ferroan alkali-calcic metaluminous rhyolite) are consistent with post-collisional geodynamic settings described in literature (e.g. Sylvester 1989). Following the classification of A-type granites of Grebennikov (2014), and considering a scheme of the origin of A-type granitoids suggested by Frost and Frost (2011), the rhyolite studied shows features transitional between the A1-type and A2-type granites. The relatively low Y/Nb ratios of 1.1 and 2.1, and high Cr contents (up to 90 ppm) indicate that the rock may represent a differentiate of a mantle-derived, alkali basalt magma (e.g. Eby 1992; Frost and Frost 2011). As shown by Eby (1992), crustal contamination of A1 group granitic magmas may increase Y/Nb ratio so that it plots in the A2 field.

The negative Ti, Ba, Sr, P and Eu anomalies in the multielement plot (Fig. 6a) can be interpreted as a result of fractionation of feldspars, apatite and Ti-rich minerals. The lanthanide tetrad effect is usually reported from highly evolved granites (Bau 1996) and indicates a fluid/melt interaction within this system (Irber 1999). According to Yang et al. (2014), formation of hydrothermal and metamict zircons can be responsible for the origin of the M-type tetrad effect. Fluid interaction in the late- or post-magmatic phase is also supported by secondary enrichment of the alkali amphibole with Zr. Increased oxygen fugacity during the terminal phases of hydrothermal process is indicated by crystallization of hematite in miarolitic cavities, whereas the rock itself contains accessory ilmenite.

The conventional K–Ar geochronology of the sample studied yielded 299.9 ± 9.1 Ma, indicating latest Carboniferous or Early Permian igneous activity. A recently studied Early Permian peralkaline rhyolite from southern Altaids exhibits similar petrology, chemical composition and tectonic setting, and a very

### Tab. 3 Chemical composition of astrophyllite (wt. %), normalization scheme based on 26 O, 5 (OH + F)

| Element | I | II | III | II | II |
|---------|---|----|-----|----|----|
| NbO₅    | 6.44 | 5.32 | 4.76 | 2.51 | 4.09 |
| SiO₂    | 33.61 | 33.88 | 33.65 | 34.90 | 34.63 |
| TiO₂    | 5.10 | 4.74 | 4.90 | 8.22 | 7.24 |
| ZrO₂    | 3.45 | 4.90 | 5.00 | 2.55 | 2.07 |
| Al₂O₃   | 1.35 | 1.24 | 1.10 | 0.97 | 1.17 |
| V₂O₅    | 0.03 | 0.02 | 0.04 | 0.06 | 0.05 |
| Cr₂O₃   | 0.01 | 0.01 | 0.01 | 0.00 | 0.00 |
| MgO     | 0.02 | 0.00 | 0.01 | 0.01 | 0.01 |
| CaO     | 0.24 | 0.24 | 0.45 | 0.69 | 0.36 |
| MnO     | 1.66 | 1.50 | 1.46 | 1.46 | 1.63 |
| FeO⁺    | 34.57 | 33.84 | 33.93 | 33.79 | 35.16 |
| ZnO     | 0.36 | 0.28 | 0.34 | 0.32 | 0.41 |
| Na₂O    | 2.01 | 2.18 | 2.11 | 2.00 | 2.14 |
| K₂O     | 5.82 | 5.78 | 6.03 | 5.48 | 5.55 |
| RbO     | 0.60 | 0.55 | 0.59 | 0.47 | 0.53 |
| Cs₂O    | 0.11 | 0.06 | 0.08 | 0.05 | 0.03 |
| F       | 0.89 | 0.91 | 0.92 | 1.06 | 0.97 |
| Cl      | 0.02 | 0.00 | 0.02 | 0.00 | 0.02 |
| H₂O     | 2.66 | 2.28 | 2.49 | 1.94 | 2.27 |
| -O=F    | 0.38 | 0.38 | 0.39 | 0.45 | 0.41 |
| Total   | 98.56 | 97.36 | 97.47 | 96.03 | 97.93 |
| Nb      | 0.662 | 0.545 | 0.492 | 0.252 | 0.410 |
| Si      | 7.640 | 7.668 | 7.704 | 7.745 | 7.693 |
| Ti      | 0.872 | 0.807 | 0.845 | 1.372 | 1.210 |
| Zr      | 0.383 | 0.541 | 0.558 | 0.276 | 0.224 |
| Al      | 0.360 | 0.332 | 0.296 | 0.255 | 0.307 |
| V       | 0.005 | 0.003 | 0.008 | 0.011 | 0.009 |
| Cr      | 0.002 | 0.002 | 0.001 | 0.000 | 0.000 |
| Mg      | 0.008 | 0.000 | 0.002 | 0.003 | 0.004 |
| Ca      | 0.058 | 0.057 | 0.110 | 0.164 | 0.086 |
| Mn      | 0.320 | 0.287 | 0.282 | 0.274 | 0.307 |
| Fe      | 6.572 | 6.406 | 6.497 | 6.272 | 6.530 |
| Zn      | 0.061 | 0.047 | 0.057 | 0.052 | 0.068 |
| Na      | 0.885 | 0.958 | 0.936 | 0.859 | 0.921 |
| K       | 1.687 | 1.670 | 1.762 | 1.550 | 1.573 |
| Rb      | 0.087 | 0.080 | 0.087 | 0.067 | 0.075 |
| Cs      | 0.011 | 0.006 | 0.008 | 0.005 | 0.003 |
| Total   | 19.613 | 19.408 | 19.645 | 19.158 | 19.422 |
| F       | 0.640 | 0.654 | 0.663 | 0.746 | 0.679 |
| Cl      | 0.007 | 0.000 | 0.007 | 0.000 | 0.008 |
| OH      | 4.028 | 3.438 | 3.796 | 2.865 | 3.359 |
| O       | 26 | 26 | 26 | 26 | 26 |
| Fe/(Fe+Mn) | 0.954 | 0.957 | 0.958 | 0.958 | 0.955 |

### Tab. 4 Analytical data for the K–Ar age determination

| Analytical N° | Sample N° | K (wt. %) | 40Ar_initial (ccSTP/g) | 40Ar_total (%) | K–Ar age (Ma) error (Ma, 1σ) |
|---------------|-----------|-----------|------------------------|----------------|-----------------------------|
| 8477          | GZ195     | 3.555     | 4.5086 × 10⁻⁵          | 90.4           | 299.9 ± 9.1                 |
similar age (286.5 ± 2.1 and 286.7 ± 2.1 Ma – Mao et al. 2014).

The presence of A1-type volcanism gives evidence of early Permian extension in the western Mongolian Altai, which can be related to initial rifting (see e.g. Eby 1992; Bonin 2007).

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Fig. 6 Multi-element diagrams. a – Primitive mantle normalized pattern. b – Chondrite-normalized REE pattern. Normalizing values are taken from McDonough and Sun (1995) and Boynton (1984), respectively.

Fig. 7 Whole-rock geochemical discrimination diagrams. Ce (a) and Y (b) vs. 10000 × Ga/Al after Whalen et al. (1987), showing the fields of I-, S-, and A-type granites. c – Ternary plot Fe2O3t × 5 – (Na2O + K2O) – (CaO + MgO) × 5 for classification of A-type (ferroan) granites, (after Grebennikov 2014). d – Ternary plot Y – Nb – 3 × Ga for classification of A-type granites (Eby 1992).
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