Enhanced charge separation in TiO$_2$/nanocarbon hybrid photocatalysts through coupling with short carbon nanotubes†

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The interfacial contact between TiO$_2$ and graphitic carbon in a hybrid composite plays a critical role in electron transfer behavior, and in turn, its photocatalytic efficiency. Herein, we report a new approach for improving the interfacial contact and delaying charge carrier recombination in the hybrid by wrapping short single-wall carbon nanotubes (SWCNTs) on TiO$_2$ particles (100 nm) via a hydration-condensation technique. Short SWCNTs with an average length of 125 ± 90 nm were obtained from an ultrasonication-assisted cutting process of pristine SWCNTs (1–3 μm in length). In comparison to conventional TiO$_2$–SWCNT composites synthesized from long SWCNTs (1.2 ± 0.7 μm), TiO$_2$ wrapped with short SWCNTs showed longer lifetimes of photogenerated electrons and holes, as well as a superior photocatalytic activity in the gas-phase degradation of acetaldehyde. In addition, upon comparison with a TiO$_2$–nanographene ‘quasi-core–shell’ structure, TiO$_2$–short SWCNT structures offer better electron-capturing efficiency and slightly higher photocatalytic performance, revealing the impact of the dimensions of graphitic structures on the interfacial transfer of electrons and light penetration to TiO$_2$. The engineering of the TiO$_2$–SWCNT structure is expected to benefit photocatalytic degradation of other volatile organic compounds, and provide alternative pathways to further improve the efficiency of other carbon-based photocatalysts.

1. Introduction

Rapid recombination of photogenerated electron–hole pairs is one of the major limitations of utilizing TiO$_2$ photocatalysts in applications such as water treatment,¹–³ air purification⁴–⁹ bacterial and viral inactivation,¹⁰¹¹ and self-cleaning glasses.¹²–¹⁴ Most of the photogenerated electron–hole pairs recombine within 10 picoseconds (ps), which is very short compared to the 100 ps typically required for redox reactions. As a result, only a small portion (~10 percent) of the photogenerated charges, as revealed by time-resolved absorption spectroscopic studies, remain alive and able to participate in pollutant decomposition, resulting in limited photocatalytic activity.¹⁵,¹⁶ A number of studies have shown that inhibition of charge recombination and remarkable enhancement in photocatalytic efficiency of TiO$_2$ can be achieved by coupling with carbon nanomaterials such as carbon nanotubes (CNTs), graphene, and fullerene.¹⁷–²¹

The coupling leads to formation of a Schottky barrier, through which electrons efficiently travel from TiO$_2$ with a higher Fermi level to the carbon nanomaterial with a lower Fermi level until equilibrium is attained.²²,²³ In addition, the high electron-storage capacity of the graphitic structure (one electron per 32 carbon atoms) can serve as electron reservoirs or sinks, decreasing the recombination probability of the photogenerated charges in the hybrid material and enhancing photocatalytic activity.²²

The improved charge separation in TiO$_2$–nanocarbon hybrid structures depends largely on the quality of interfacial contact between TiO$_2$ and the graphitic structure, and the structural and surface properties of the hybrid components.²¹,²⁴–³⁰ A knowledge gap that still exists is how the graphitic structures and their geometries affect the coupling and interfacial contacts of the components in the hybrid. The rate of electron transport and subsequent separation of photogenerated charges in conventional TiO$_2$–nanocarbon composites are still hampered by nonuniform coating or poor dispersion induced by TiO$_2$ aggregation on graphene sheets or CNT walls. Therefore, in light of the great promise shown by TiO$_2$–nanocarbon hybrids in photocatalysis, research efforts have focused on tailoring the heterojunctions and schottky barriers to achieve efficient shuffling of electrons and promote electron–hole separation, as well as induce visible light excitation. It is well-established in the literature that charge recombination is inhibited on
a variety of TiO$_2$–CNT hybrids consisting of a thin, uniform TiO$_2$ layer on individual multi-walled CNTs (MWCNTs) or TiO$_2$ aggregates supported on CNTs. However, the trade-off relationship between electron–hole concentration and separation efficiency in conventional TiO$_2$-coated CNT composites remains a significant impediment to maximizing the photocatalytic activity. To this end, TiO$_2$ wrapped with graphene has been considered as an alternative to the TiO$_2$/CNT core–shell (or TiO$_2$-coated CNT) structure, as the concentration of photogenerated electrons and holes, as well as a superior photocatalytic activity in the gas-phase degradation of acetaldehyde. In addition, upon comparison with a TiO$_2$-nanographene “quasi-core–shell” structure, the novel TiO$_2$–SWCNT composite offers superior electron-capturing efficiency and slightly higher photocatalytic performance, revealing the impact of the dimension of graphitic structures on the interfacial transfer of electrons and light penetration to TiO$_2$.

2. Experimental section

2.1. Synthesis of TiO$_2$–SWCNT composites

Purified SWCNTs (1–2 nm in diameter and 1–3 μm in length) were functionalized by refluxing with HNO$_3$ (70%) at 150 °C for two hours. A known amount of functionalized SWCNTs were dispersed in ethanol and sonicated at 25 °C for four hours to produce the short SWCNTs. After the sonication-assisted cutting process, the resulting suspension was allowed to settle for 15 minutes and the supernatant was slowly added to the TiO$_2$–ethanol dispersion (50 mg in 50 mL ethanol) under sonication, followed by robust stirring overnight at pH = 3.4. The sonicator (Branson, CPX5800H) operated at 100 W and 40 kHz. The TiO$_2$ particles (anatase, 99.9%, 100 nm) were purchased from US Research Nanomaterials, Inc. The TiO$_2$–SWCNT hybrid synthesized from short SWCNTs (hereafter referred to as TiO$_2$–short SWCNT) was collected by filtration using a membrane with average pore size of 0.2 μm and dried at 60 °C. For TiO$_2$–SWCNT hybrids synthesized from long SWCNTs, the same procedure was followed without subjecting the long SWCNTs to ultrasonication-assisted cutting; the resulting hybrid is hereafter referred to as TiO$_2$–long SWCNT. Unless stated otherwise, the amount of SWCNTs used in the standard sample is 1 wt%.

2.2. Synthesis of TiO$_2$–nanographene composites

The graphite powder was oxidized by the modified Hummer method. The graphene oxide (GO) produced was treated with a mixture of HCl and KMnO$_4$ to produce the nanographene oxide. GO (0.05 g) was dispersed in 50 mL of concentrated HCl, and then 0.05 g of KMnO$_4$ was slowly added under robust stirring. After one hour of stirring at 40 °C, the resulting slurry was cooled in an ice bath. Then, H$_2$O$_2$ solution (5 mL H$_2$O$_2$ and 95 mL distilled water) was slowly added with continued stirring for one hour, followed by sonication for 30 minutes to exfoliate the nanographene oxide. The final product was collected by filtration using similar conditions in Section 2.1 and oven-dried at 60 °C.

The TiO$_2$–nanographene composite (0.7 wt% nanographene) was synthesized by dispersing a known amount of nanographene oxide in ethanol by sonication, and the resulting slurry was slowly added to the TiO$_2$–ethanol dispersion (50 mg in 50 mL distilled water) under stirring at pH = 3.5, followed by robust stirring overnight. The brownish powder obtained was transferred to a Teflon-lined, hydrothermal synthesis autoclave reactor and heated at 120 °C for 24 hours to reduce the nanographene oxide. The final product (hereafter referred to as TiO$_2$–nanographene) was washed repeatedly with 1 M HCl and distilled water, collected by filtration and dried at 60 °C.
2.3. Characterization methods

X-ray diffraction (XRD) patterns were collected on a Rigaku Miniex II X-ray diffractometer with Cu Kα radiation (λ = 1.5418 Å) at 40 kV and 40 mA. The Raman spectra were acquired via an iHR550 spectrometer with a laser excitation wavelength of 532 nm. Catalyst morphology was characterized by a field-emission scanning electron microscope (SEM, Hitachi S5200) equipped with energy-dispersive X-ray spectroscopy (EDS), and a transmission electron microscope (TEM, FEI Talos) operating at 200 kV. For TEM analysis, a small amount of the sample was dispersed in ethanol by sonication and dropped onto a copper microgrid coated with lacy carbon film. X-ray photoelectron spectroscopic (XPS) measurements were collected via a Kratos Ultra XPS system at 1.0 × 10⁻⁹ Torr. A monochromatic Al Kα source (hν = 1486.6 eV) operating at 10 mA and 12 keV (120 W) was used to collect the spectra. The effect of charging was reduced by a charge neutralizer. Survey spectra were collected at 160 eV pass energy with 1 eV step and 100 ms contact time. For the high-resolution scans, spectra were collected at 20 eV pass energy with 0.1 eV step and 100 ms contact time. The photoluminescence (PL) spectra were acquired via a spectraMax i3x multi-mode microplate reader over the range of 400–700 nm using an excitation wavelength of 360 nm.

2.4. Evaluation of photocatalytic activity

The photocatalytic activity was tested by measuring the change in concentration of gas phase acetaldehyde as a function of irradiation time. The photocatalyst (100 mg) was placed in a 750 mL borosilicate batch reactor surrounded by light sources (two 25 W Sylvania 21703 (356 nm) bulbs and two 13 W ReptiSun 10.0 UV/B bulbs (280–320 nm) with intensity of 6.5 W m⁻²). One thousand ppmv of acetaldehyde gas was injected into the reactor via a syringe. After equilibrium between gaseous and adsorbed phases was attained, UV irradiation was started and the decrease in concentration was measured via an SRI 310C gas chromatograph equipped with a Restek silica gel column (8046-895, 6 ft length, 2 mm ID), on-column injection, and flame ionization detector (FID). According to the experimental data, acetaldehyde degradation by TiO₂-short SWCNT, TiO₂-long SWCNT and TiO₂–nanographene fitted to a pseudo first-order reaction represented by eqn (1):

\[
\frac{c}{c_0} = e^{-kt}
\]

where \( k \) is the apparent constant of reaction rate, \( c \) is the concentration of acetaldehyde and \( t \) is the reaction time.

3. Results

3.1. Morphology, structure, composition, and interfacial properties

The SEM image in Fig. 1a reveals TiO₂ particles are partially wrapped by short SWCNTs (TiO₂-short SWCNT), a phenomenon that occurs uniformly over a large area on TiO₂ particles. EDS mapping of TiO₂-short SWCNT (Fig. 1b) reveals the co-existence

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**Fig. 1** (a) SEM image of TiO₂-short SWCNT; (b) EDS mapping of TiO₂-short SWCNT; (c) and (d) TEM images of (a) showing TiO₂ particles decorated with short SWCNTs and SWCNT bundles with arrows showing interactions between a TiO₂ particle and individual SWCNTs or bundles; (e) A zoom-in high-resolution TEM image of the region in (d) with a red square.
of carbon, titanium, and oxygen, with atomic concentrations of 15.75%, 34.23%, and 32.62%, respectively. The partial wrapping of TiO₂ particles by individual SWCNTs or SWCNT bundles is further confirmed by the TEM images in Fig. 1c and d while the low-magnification SEM image in Fig. S1† demonstrates the wrapping phenomenon occurs across a large area of the sample. The TEM images (Fig. 1c and d) show short SWCNTs of different lengths either as bundles or isolated nanotubes on a TiO₂ particle. From SEM and TEM data, it is apparent intimate interactions between the SWCNTs and TiO₂ exist. The high-resolution TEM images in Fig. S2† show the SWCNTs form a network on TiO₂ particles. The small diameters of SWCNTs (~1.5 nm) compared to that of TiO₂ particles (~100 nm) implies a large fraction of the incident light can participate in excitation of TiO₂. In other words, the deposition of the short SWCNTs does not appear to lead to significant light scattering or prevent light absorption by TiO₂. However, aggregation of the short SWCNTs was observed in photocatalyst samples with higher SWCNT concentrations as shown in the TiO₂-short SWCNT sample with 3 wt% SWCNTs (Fig. S3†).

XRD patterns of short SWCNTs, TiO₂, and TiO₂-short SWCNT are shown in Fig. 2a. The peaks at 2θ = 55.1°, 54.0°, 48.1°, 37.9°, and 25.4° are respectively indexed to (211), (105), (200), (004), and (101) facets, typical for anatase TiO₂. The sharp and intense peaks affirm the high crystallinity of the photocatalysts, whereas the absence of peaks associated with impurities is evidence of their high quality. The peaks at 26.8° and 42.8°, usually attributed to (002) and (100) crystal planes of SWCNTs, were not detected in TiO₂-short SWCNT, due to the low SWCNT concentration (1 wt%). However, the addition of SWCNTs slightly reduced the relative intensity of TiO₂ peaks, an observation consistent with other studies. Raman spectra (Fig. 2b) show the characteristic peaks of anatase TiO₂ at 397, 515, and 639 cm⁻¹. The characteristic anatase peaks (B₁g, A₁g + B₁g, and E₂g) were observed in TiO₂-short SWCNT, accompanied by the disordered carbon-induced D-band at 1350 cm⁻¹, and the G-band at 1583 cm⁻¹ that represents sp² carbon from SWCNTs. The slight shift in anatase modes observed in the Raman and XRD spectra generally correlates with the external perturbation by SWCNTs and the corresponding substitution of Ti⁴⁺ by C atoms, resulting in the formation of Ti–O–C bonds in the TiO₂ lattice. However, band gaps of TiO₂ and TiO₂-short SWCNT (calculated from UV-Vis spectra in Fig. S4†) are 3.22 and 3.19 eV, respectively, suggesting low-carbon doping in the composite.

An XPS survey spectrum of TiO₂-short SWCNT confirms the presence of Ti, O, and C (Fig. 3a). The high-resolution XPS spectrum of Ti 2p (Fig. 3b) shows two peaks located at 459.5 and 465.2 eV attributed to Ti 2p3/2 and Ti 2p1/2, respectively, which are typical binding energies of the Ti⁴⁺ valence state. The O 1s spectrum, shown in Fig. 3c, is fitted with three peak components. The component at 530.7 eV corresponds to Ti–O–Ti, while the peak at 531.6 eV is attributed to C=O or C–OH. The peak at 532.8 eV is ascribed to Ti–O–C, and is an indication of the covalent interaction between TiO₂ and SWCNTs. Based on density functional theory calculations, Ti–O–C bonds facilitate charge carrier separation due to their higher electrostatic density in comparison to the non-covalent interaction. As a consequence, the charge redistribution map predicts a higher rate of electron transport. The C 1s region, shown in Fig. 3d, is...
fitted with four peak components. The peak at 284.5 eV is assigned to sp² carbon, while the peaks at 285.7, 287.7, and 289.8 eV are assigned to functional groups linked to the carbon atoms including C-O/C-O-Ti, C=O, and O–C==O, respectively.⁵⁴,⁵⁶ The various oxygen functional groups generate anchors for direct linking with TiO₂. The absence of the Ti=C peak, typically located at 281 eV,⁵⁷ confirms our claim regarding the absence of significant carbon doping in the composite.

### 3.2. Photocatalytic degradation of gaseous acetaldehyde by TiO₂–SWCNT hybrids

To evaluate the photocatalytic activity of TiO₂ wrapped with SWCNTs, the degradation of acetaldehyde was conducted under UV light. Fig. 4a shows the effect of SWCNT amount in the hybrid structure on photocatalytic activity expressed in terms of normalized acetaldehyde concentration a as a function of the amount of SWCNTs in the composite; our findings revealed the optimum amount to be 1 wt%. A schematic illustration of the proposed mechanism for photocatalytic degradation of gaseous acetaldehyde on TiO₂–short SWCNT hybrid is depicted in Fig. 4b. Upon light illumination, photo-induced electron–hole pairs are produced by TiO₂ (hv (<350 nm) → e⁻ CB+SWCNTs + h⁺); the electrons are promoted to the conduction band leaving holes in the valance band.⁵⁸ The CB position of anatase is ~4.21 eV using vacuum level (AVS) (~0.4 eV relative to reversible hydrogen electrode (RHE)), with a bandgap of about 3.2 eV. On the other hand, the work function of SWCNTs is known to be ~4.8 eV (AVS) (approximately 0.2 eV relative to RHE).⁵⁹ Since the work function of SWCNTs is lower than the conduction band of TiO₂, SWCNTs can act as electron sinks whereby a fraction of the photoelectrons flow to them.⁶⁰ The holes in the valance band react with water molecules to generate hydroxyl radicals (‘OH) (H₂O + h⁺ → ‘OH + H⁺), while the photoexcited electrons in SWCNTs and in the conduction band of TiO₂ react with O₂ to produce superoxide radicals (‘O₂⁻) (O₂ + e⁻ CB+SWCNTs → ‘O₂⁻). The active radicals initiate decomposition of adsorbed gaseous acetaldehyde via two reaction pathways. ‘OH and ‘O₂⁻ could oxidize a portion of acetaldehyde to carbon dioxide and water (CH₃CHO + ‘OH → CO₂ + H₂O). The remaining acetaldehyde could be oxidized to acetic acid by ‘OH in the first step (CH₃CHO + ‘OH → CH₃COOH) and subsequently converted to water and carbon dioxide by ‘O₂⁻ in a second step (CH₃CHO + ‘O₂⁻ → CO₂ + H₂O).⁶¹,⁶² As shown in Fig. 4c, the relevant potential level of the acceptor species is located below the CB potential of TiO₂. On the other hand, the potential level of the donor species is above the valance band position of TiO₂, validating the generation of active radicals responsible for the degradation of gaseous acetaldehyde.⁶³–⁶⁹

It is well known the photogenerated electron–hole pair participates in redox reactions on the catalyst surface to initiate degradation of adsorbed pollutant; however, the process is limited by rapid recombination of electrons and holes. As depicted in Fig. 4b, one of the roles of SWCNTs in TiO₂–CNT composites is to inhibit electron–hole recombination.¹⁵,¹⁶ An increase in the amount of SWCNTs in the hybrid composite beyond 1 wt% has a negative effect on the photocatalytic performance, which we attribute to SWCNT aggregation and generation of large SWCNT bundles. Hsu and coworkers⁶⁷–⁷⁵ have provided further mechanistic insight into the effect of nanocarbon loading on photocatalytic performance. The overall charge separation efficiency in nanocarbon-semiconductor hybrids can be sacrificed when the amount of the electron sink (metal or nanocarbon) exceeds the optimum value. The observed trend in activity for the different TiO₂–SWCNT ratios in our study is consistent with the results from these studies and can also be rationalized by the so called later-emerging electron–hole recombination that is induced across the interface. The formation of large-diameter bundles indicates SWCNTs at high concentration tend to interact with each other via Van der Waals forces rather than with TiO₂. As a consequence, the quality of the interfacial contact (or degree of interaction) between TiO₂ and SWCNTs is reduced. In addition, the coverage of thick SWCNT bundles on TiO₂ particles has the potential of significantly decreasing light penetration to TiO₂,
thus reducing photoactivity as observed in other studies that involve wrapping of graphene on TiO$_2$ or coupling graphene quantum dots with TiO$_2$.\textsuperscript{41,76} Despite the decrease in photocatalytic activity upon SWCNT aggregation, the observed performance is still higher than pristine TiO$_2$, indicating photon permeation is not completely blocked and the beneficial role of SWCNTs in photocatalytic enhancement, via charge separation, may still be partially maintained.

3.3. Effect of SWCNT length and graphitic dimensions on interfacial contact, charge separation, and photocatalytic activity of the hybrids

To probe the influence of SWCNT length on the interfacial contact and subsequent photocatalytic performance of TiO$_2$–SWCNT hybrids, we synthesized another composite with long SWCNTs (TiO$_2$-long SWCNT) using the same concentration of SWCNTs, but without subjecting the sample to the ultrasonication cutting step. Representative TEM images for long SWCNTs and short SWCNTs used for coupling with TiO$_2$ are presented in Fig. S5 with corresponding box plots in Fig. 5 showing their length distributions; each box plot consists of 12 measurements of nanotubes from TEM images. The average length of short SWCNTs is $125 \pm 90$ nm while that of long SWCNTs is $1.2 \pm 0.67$ \(\mu\)m. The results clearly demonstrate the shear force generated by ultrasonication results in SWCNTs that are significantly shorter than those processed without ultrasonication. The use of ultrasonication as a nondestructive technique for cutting nanotubes is well established in processing nanomaterials. Shuba et al.\textsuperscript{77} showed that ultrasonication yields SWCNTs with low side-wall degradation and preserves the electronic properties of the nanotube. Note that for SWCNTs to efficiently play the intended role of “electron sink” in hybrid structures during photocatalysis, preserving their structural integrity and exceptional electronic properties is critically important.

The Raman spectra of pristine SWCNTs, short SWCNTs, and long SWCNTs in Fig. 6 provide further insight into the effect of the processing approach on the structure of SWCNTs. The ratio of G- to D-band intensities ($I_G/I_D$) from the Raman spectra, widely considered a quality or graphitization index for carbon-based materials,\textsuperscript{78–80} is 13.9 for pristine SWCNTs and decreases to 3.3 and 3.6 after acid treatment for long SWCNTs and short SWCNTs, respectively, due to the functionalization of SWCNTs. The $I_G/I_D$ ratios for short and long SWCNTs are about the same, which indicates that the ultrasonication process did not degrade the nanotubes. As observed previously in Fig. 2b, the TiO$_2$–SWCNT composite exhibited a slight decrease in $I_G/I_D$, suggesting the increase in defect after the coupling of TiO$_2$ and SWCNTs,\textsuperscript{49} probably due to the low substitution of Ti$^{4+}$ by carbon atoms. In addition, the characteristic phonon mode in SWCNTs occurring in the low frequency region (zoomed in the insets), the radial breathing mode (RBM), is clearly observed in pristine and processed SWCNTs. In fact, the same RBM peaks for pristine SWCNTs at $150$ and $260$ cm$^{-1}$ are observed for long SWCNTs and short SWCNTs. This provides further evidence that the ultrasonication process did not destroy the nanotubes.

Next, we investigate the effect of SWCNT length and nanocarbon dimension on interfacial contact, charge separation, and photocatalytic activity of the hybrids. Electron microscopic images of the resulting hybrids formed with short SWCNTs, long SWCNTs and nanographene, along with corresponding EDS mapping for TiO$_2$–nanographene, are shown in Fig. 7. The EDS mapping of TiO$_2$–nanographene reveal a highly uniform distribution of titanium, carbon, and oxygen as observed for TiO$_2$-short SWCNT (Fig. 1b). In contrast to TiO$_2$-short SWCNT (Fig. 7a), bundles of large-diameter SWCNTs are apparent in TiO$_2$-long SWCNT composites (Fig. 7b), with severe particle aggregation as shown in Fig. S6.\textsuperscript{†} These observations give credence to the assumption that ultrasonication not only reduces the length of SWCNTs, but also facilitates debundling of SWCNTs. Morphological characterization of micron-sized
Graphene, exfoliated nanographene, and non-exfoliated nanographene are summarized in Fig. S7.† The average size the nanographene, as determined from TEM analysis and shown on the box plot (Fig. S7†), is \( \sim 125 \pm 93 \) nm.

Photocatalytic degradation of acetaldehyde by TiO\(_2\), TiO\(_2\)-long SWCNT, TiO\(_2\)-nanographene, and TiO\(_2\)-short SWCNT, is shown in Fig. 8a along with the PL spectra of the photocatalysts in Fig. 8b. As depicted in Fig. 8a, TiO\(_2\)-short SWCNT shows the highest performance; the rate constant of TiO\(_2\)-short SWCNT is \( 0.015 \text{ min}^{-1} \), while others are given as follows: pristine TiO\(_2\) (\( k = 0.0073 \text{ min}^{-1} \)), TiO\(_2\)-long SWCNT (\( k = 0.0107 \text{ min}^{-1} \)), and TiO\(_2\)-nanographene (\( k = 0.0125 \text{ min}^{-1} \)). TiO\(_2\)-short SWCNT showed better performance compared with TiO\(_2\)-nanographene, even though the quality of interfacial contact in the two heterostructures is somewhat similar, as evidenced by EDS mapping that reveals a highly uniform distribution of nanographene on TiO\(_2\) particles.

Fig. 7  SEM images of TiO\(_2\)-short SWCNT (a) and TiO\(_2\)-long SWCNT (b). TEM image of TiO\(_2\)-nanographene (c) and corresponding EDS mapping (d). TEM image of TiO\(_2\)-short SWCNT (e).

Fig. 8  (a) Photocatalytic activity of TiO\(_2\), TiO\(_2\)-short SWCNT and TiO\(_2\)-long SWCNT during degradation of gaseous acetaldehyde based on a pseudo first-order kinetic model. (b) Photoluminescence (PL) spectra of TiO\(_2\), TiO\(_2\)-long SWCNT, TiO\(_2\)-nanographene, TiO\(_2\)-short SWCNT, and short SWCNT.
over a large area (Fig. 7d). The PL spectra (Fig. 8b) show lifetimes of charges on TiO$_2$ coupled with short SWCNTs are longer than TiO$_2$ coupled with nanographene, which explains the different photocatalytic activities observed as evidenced by their rate constants.

4. Discussion

Based on our results, we present a schematic illustration of the interfacial contact between nanocarbon (short SWCNTs, long SWCNTs or nanographene) and TiO$_2$, as well as the charge carrier movement in the hybrids in Fig. 7. We rationalize the superior performance of TiO$_2$-short SWCNT to be due to the inhibition of charge recombination in the composite, which can be achieved in two ways. First, the increase in the number of interfacial contacts between each individual TiO$_2$ and short SWCNTs provides more channels for electron transport that inhibits charge recombination. The short and well-exfoliated SWCNTs can have lower bending stiffness compared to the long and non-exfoliated SWCNTs. It has been shown that bending stiffness of SWCNTs is directly proportional to SWCNT diameter, therefore, we expect large-diameter SWCNT bundles to have larger persistence length compared to exfoliated SWCNTs (or small-diameter SWCNT bundles). Hence, short SWCNTs efficiently wrap a TiO$_2$ particle (Fig. 9a). Consequently, a large fraction of TiO$_2$ nanoparticles are in direct contact with short SWCNTs. In contrast, only a small fraction of TiO$_2$ nanoparticles are in direct contact with the long SWCNTs (Fig. 9b). Second, the improved SWCNT debundling ensures a uniform distribution of SWCNTs over a large area on the TiO$_2$ surface, and hence, the shuttling of electrons across the interface for enhanced charge separation is not restricted to a small portion of the composite as in the case of TiO$_2$-long SWCNT hybrids. To test the above hypotheses, PL data presented in Fig. 8b provide valuable insight. In comparison to pristine TiO$_2$, fluorescence quenching was observed in the TiO$_2$-SWCNT hybrid structures, confirming the transfer of electrons from TiO$_2$ to SWCNTs delayed charge recombination. It is clear from the PL results that the intensity of TiO$_2$-short SWCNT hybrids is lower than that of TiO$_2$-long SWCNT hybrids, and is substantially lower than pristine TiO$_2$, confirming the efficient charge separation occurring in TiO$_2$-short SWCNT hybrids.

To rationalize the role of short SWCNTs (or oxygen-containing functionalities), we have to consider the synthesis steps, and in particular, the zeta potentials of SWCNTs and TiO$_2$ particles in ethanol under the pH used. Factors such as type of solvent and pH of mixture can alter the surface charge of nanostructures. In our case, acid treatment lowers the zeta potential of SWCNTs due to the formation of oxygen functionalities as shown by Liu et al. They showed that the addition of a relatively small amount of acid or base is capable of introducing a relatively strong charge on TiO$_2$ particles. In our synthesis experiment, TiO$_2$ particles and functionalized SWCNTs were dispersed in ethanol at pH $= 3.5$ which creates an environment with a high hydrogen ion activity, the surface of TiO$_2$ acquires a positive charge due to the high zeta potential. In fact, Widegren and Bergström showed that the addition of a relatively small amount of acid or base is capable of introducing a relatively strong charge on TiO$_2$ particles. In our synthesis experiment, TiO$_2$ particles and functionalized SWCNTs were dispersed in ethanol at pH $= 3.5$ which creates an environment with a high hydrogen ion activity, and a strong positive charge on TiO$_2$. The strong positively charged TiO$_2$ particles and negatively charged functionalized SWCNTs interact strongly via electrostatic attraction to form the hybrid structures with enhanced interfacial contact. In addition, opened and unblocked CNTs significantly increase the density of adsorption or anchoring sites. Cutting of long SWCNTs increases the number of open ends, and therefore, the short SWCNTs provide more “handles” or anchoring sites for interactions with TiO$_2$ than long SWCNTs.

The dimension of the graphitic structure is another important factor that can affect charge separation. To understand the role of the dimension of the graphitic structure in the composite, we compared the performance of TiO$_2$ quasi-wrapped by 1D SWCNTs (TiO$_2$-short SWCNT) with the quasi-wrapped by 2D nanographene (TiO$_2$-nanographene). The synthesis of the quasi-core–shell structure is adapted from a technique proposed by Kim et al. We hypothesize that random pathways in 2D graphene can significantly reduce the...
electron transport rate or may lead to electron escape from the graphene sheet (as schematically depicted in Fig. 9c), and thus promote charge carrier recombination. In contrast, the 1D SWCNTs offer efficient and short pathways for the transfer of vectorial electrons and, hence, enhanced charge separation. In addition, nanographene has a high stacking tendency and even after exfoliation, the number of layers cannot be easily controlled. During TEM analysis, we observed several TiO2 nanoparticles with thick graphic shells, considerably thicker than SWCNT bundles (see Fig. 7c and e). This means significant portion of incident light is hindered from reaching TiO2 in a typical core–shell structure for TiO2–nanographene. On the other hand, SWCNTs, even in the case of bundles, due to their one-dimensional structure and small diameters, do not completely block light permeation to TiO2, as shown earlier for hybrid structures synthesized using high SWCNT concentrations. Nevertheless, coupling short SWCNTs or nanographene significantly improved electron–hole separation and photoreactivity of TiO2; the photoexcited electrons in the TiO2 core are efficiently transported into the nanocarbon due to the strong interfacial contact between the hybrid components.

5. Conclusions

TiO2 nanoparticles quasi-wrapped with short SWCNTs were successfully fabricated via an approach that involves the interaction between TiO2 and short SWCNTs in ethanol as a solvent. The presence of short SWCNTs resulted in enhanced photocatalytic activity and efficient charge separation. The improved interfacial contact, verified by electron microscopy, prolonged the lifetime of an electron–hole pair on TiO2 as demonstrated by photoluminescence spectroscopy. Unlike hybrid structures with long SWCNTs, higher photocatalytic performance in gas-phase degradation of acetaldehyde was achieved with TiO2–short SWCNT hybrids, confirming use of short SWCNTs in coupling with TiO2 as a better alternative. Investigations into the effect of nanocarbon dimensions on charge separation revealed TiO2–short SWCNT hybrids have slightly higher photocatalytic activity than TiO2–nanographene. We attribute the performance to electron scattering in 2D-nanographene, which may affect electron transport through the interface. The high stacking tendency of nanographene may form a thick graphic layer on TiO2 that could hinder UV light penetration. Overall, this study provides rational new guidelines for maximizing electron–hole separation in TiO2–nanocarbon hybrids for improved photocatalytic performance.

Conflicts of interest

There are no conflicts to declare.

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