Excited-State Dynamics in Colloidal Semiconductor Nanocrystals

Freddy T. Rabouw1,2,3 · Celso de Mello Donega4

Received: 12 May 2016 / Accepted: 23 July 2016 / Published online: 9 August 2016
© The Author(s) 2016. This article is published with open access at Springerlink.com

Abstract Colloidal semiconductor nanocrystals have attracted continuous worldwide interest over the last three decades owing to their remarkable and unique size- and shape-, dependent properties. The colloidal nature of these nanomaterials allows one to take full advantage of nanoscale effects to tailor their optoelectronic and physical–chemical properties, yielding materials that combine size-, shape-, and composition-dependent properties with easy surface manipulation and solution processing. These features have turned the study of colloidal semiconductor nanocrystals into a dynamic and multidisciplinary research field, with fascinating fundamental challenges and dazzling application prospects. This review focuses on the excited-state dynamics in these intriguing nanomaterials, covering a range of different relaxation mechanisms that span over 15 orders of magnitude, from a few femtoseconds to a few seconds after photoexcitation. In addition to reviewing the state of the art and highlighting the essential concepts in the field, we also discuss
the relevance of the different relaxation processes to a number of potential applications, such as photovoltaics and LEDs. The fundamental physical and chemical principles needed to control and understand the properties of colloidal semiconductor nanocrystals are also addressed.

**Keywords** Semiconductor nanocrystals · Colloids · Exciton dynamics · Nanoscale · Auger relaxation

### 1 Introduction

Since the pioneering work of Brus, Ekimov, and many others in the early 1980–1990s [1–13], the study of semiconductor nanocrystals (NCs) has developed into a mature, dynamic and multidisciplinary research field, which attracts increasing attention worldwide, both for its fundamental challenges and its potential for a number of technologies (light emitting devices, solar cells, luminescent solar concentrators, optoelectronics, sensing, thermoelectrics, biomedical applications, catalysis) [14–32]. Colloidal semiconductor NCs are particularly attractive, since they consist of an inorganic core that is coated with a stabilizing layer of (usually) organic ligand molecules. This hybrid inorganic–organic nature makes them very versatile nanomaterials that combine size-, shape-, and composition-dependent optoelectronic properties of the core with easy surface manipulation and solution processing [16].

Here, we will address the excited-state dynamics in colloidal semiconductor NCs, covering a time scale that spans over 15 orders of magnitude, from a few femtoseconds to seconds after photoexcitation. We intend to provide a critical overview of the field, in which the recent advances are discussed and the outstanding challenges are identified. The relevance of different excited-state relaxation processes to a number of potential applications will also be highlighted. This review is not meant to be exhaustive, but rather to convey a concise account of the state-of-the-art, in which the essential aspects are outlined and discussed. For further details or an in-depth treatment of topics that are beyond the scope of this work, the reader will be referred to the recent literature. This article is organized as follows. In Sect. 2, we discuss how excitons in semiconductor NCs are affected by nanoscale effects. In Sect. 3, the relaxation dynamics of nanoscale excitons in colloidal semiconductor NCs is addressed, with particular emphasis on the processes that occur at different time scales after photoexcitation. In the last section, we summarize the essential aspects discussed and the outlook for the field.

### 2 Excitons in Semiconductor Nanocrystals

#### 2.1 Quantum Confinement Effects: Squeezing and Shaping Nanoscale Excitons

Absorption by a semiconductor of a photon with energy equal to or larger than its bandgap ($E_g$) promotes an electron from the valence band (VB) to the conduction
band (CB), leaving a hole in the VB, and forming an exciton (i.e., an electron–hole pair bound by Coulomb interaction). The impact of spatial confinement to the nanoscale depends on characteristic length scales associated with the physical property under consideration. In the case of the properties of excitons in semiconductors, this characteristic length scale is given by the exciton Bohr radius \( a_0 \), which ranges from \( \sim 2 \) to \( \sim 50 \) nm, depending on the material [33]. For semiconductor NC sizes of approximately \( a_0 \) and smaller, the exciton wave function is affected by spatial confinement [33]. This induces size-dependent changes in the density of electronic states and in the energy separation between them, which are manifested in an increase of the bandgap (or HOMO–LUMO energy gap) and the appearance of discrete energy levels near the band edges with decreasing NC dimensions (Fig. 1) [33–35]. This effect is commonly referred to as quantum confinement, and makes it possible to tune the optical spectra (absorption and photoluminescence, PL) of semiconductor NCs through a wide spectral window by simply changing their size, while keeping their composition constant (Fig. 1). Further, the degree of quantum confinement may be different in different directions of the NC, depending on its size and shape (Fig. 2) [33, 35]. If the exciton is

![Fig. 1](image_url)
spatially confined in all directions, a quantum dot (QD) is obtained, while NCs in which the exciton is confined only in the diameter direction are referred to as quantum wires. Quantum confinement in the thickness direction only (2D confinement) results in a quantum well. Quantum rods are NCs in transition from the zero-dimensional confinement regime of QDs to the 1D confinement regime of quantum wires. This makes the optoelectronic properties of semiconductor NCs also strongly shape-dependent. In the quantum confinement regime, the size and shape of semiconductor NCs also have an impact on the exciton fine-structure. The exciton fine-structure is the way in which the energy states of the exciton are split by effects of the crystal field asymmetry, NC shape anisotropy, and electron–hole exchange interaction [36–38]. Exciton fine-structure splitting is analogous to singlet–triplet splitting in organic molecules, but the energy splittings for an exciton in a NC are typically smaller, namely only a few meV. Effects of the exciton fine-structure are therefore relevant only at low temperatures (below 100 K), where they affect the temperature- and magnetic-field dependences of the exciton lifetimes. These effects are beyond the scope of this review. The interested reader is referred to a number of publications addressing this topic in detail [36–48].

Phonons (i.e., lattice vibrations) have a pervasive role in semiconductors, and therefore coupling of charge carriers and excitons to phonons plays a decisive role in a wide range of properties [49]. The interaction between phonons and excitons in nanoscale semiconductors is expected to differ from that in bulk materials due to both quantum confinement effects on the exciton energy levels and dimensional confinement of phonon modes (i.e., the phonon wavelength cannot be larger than the NC size) [49]. Coupling of photogenerated carriers to phonons provides an important energy relaxation pathway, thus being essential to a number of photophysical processes in semiconductor NCs (e.g., exciton relaxation dynamics, carrier cooling, thermal transport) [42, 50–53]. Moreover, coupling to acoustic phonon modes determines the homogeneous linewidths of optical transitions.
while coupling to optical phonon modes has been observed to relax selection rules at low temperatures, yielding distinct phonon-assisted transitions (the so-called phonon replicas) [56, 57].

2.2 Composition Effects: Tailoring the Property Gamut

As mentioned in Sect. 2.1 above, the exciton Bohr radius is a material property. As a result, different semiconductors experience quantum confinement at different NC sizes, depending on their exciton Bohr radius. Moreover, the bulk bandgap of different semiconductor materials covers a range of energies from the infrared to the ultraviolet. As a result, the bandgap of different semiconductor NCs is tunable over different spectral windows [16, 26, 33, 35]. For example, the lowest energy absorption transition of CdSe QDs can be tuned from 1.75 eV (the bulk $E_g$ value) to 2.65 eV for diameters ranging from $\sim 10$ nm ($a_0 = 4.9$ nm) to 2 nm [58], while that of PbSe QDs can be tuned from 0.3 eV (the bulk $E_g$ value) to 1.5 eV for diameters ranging from $\sim 100$ nm ($a_0 = 46$ nm) to 2 nm [59]. The optoelectronic properties of semiconductor NCs can thus be tailored by choosing their composition and controlling their size and shape.

The control over the properties of colloidal NCs can be extended further by using NCs consisting of two (or more) different semiconductors joined together by heterointerfaces, i.e., hetero-NCs [16]. The spatial localization of the photogenerated charge carriers in hetero-NCs can be manipulated by controlling the band offsets of the materials that are combined at the heterointerface (Fig. 3) [16]. In type-I hetero-NCs both carriers are confined in the same material (e.g., CdSe/ZnS, InP/ZnS). In contrast, in type-II hetero-NCs a spatially indirect exciton is formed, as the electron and hole wave functions are centered in different materials, and thus in different segments of the hetero-NC (e.g., CdSe/ZnTe, CdSe/CdTe). In type-I$^{1/2}$ (or quasi-type-II) hetero-NCs one carrier is delocalized over the whole volume of the hetero-NC, while the other is localized in one of the segments (e.g., CdSe/CdS, ZnSe/CdSe). This allows the electron–hole spatial overlap to be tailored by controlling the size, shape, and composition of each segment of the hetero-NC, which has a dramatic impact on several properties (viz., quantum yields, stability, PL wavelength [15, 16, 21, 60, 61], reabsorption cross section [22, 29, 62–64], radiative lifetimes [60, 64–66], exciton-phonon coupling strength [67–69], Auger recombination [66, 70–72], hot carrier relaxation [51, 73], thermal quenching [74, 75]). The general trend is that the exciton lifetime, exciton-phonon coupling, and PL wavelength increase when going from the type-I to the type-II localization regimes, while Auger recombination rates and hot carrier relaxation rates are reduced. This is beneficial not only for technologies relying on efficient charge separation, such as solar cells, photodetectors, and photocatalysis, but also for applications requiring light emission, such as luminescent solar concentrators (reduced reabsorption losses) [22, 29] and lasers (lower lasing threshold) [14]. However, short lifetimes and narrow bandwidths are preferred for application in LEDs, since this increases the output saturation threshold and the color-rendering index [19]. The electron–hole wave function overlap in hetero-NCs has also been shown to affect the exciton fine-structure [76–79].
The properties of colloidal semiconductor hetero-NCs can also be tuned under constant size, shape, and total composition by making use of controlled interdiffusion. The elemental distribution profile of hetero-NCs can go from a core/shell geometry with a sharp heterointerface to a homogenous alloy QD, via gradient alloy NCs of increasing homogeneity which seamlessly connect these two extremes [80–82]. In this way, the optoelectronic properties can be continuously tuned from those of core/shell hetero-NCs (type-I, type-II or type-I½) to those of homogeneous alloy NCs, with preservation of the total volume and composition of the NC [80–83]. Moreover, core/shell hetero-NCs with a gradient alloy heterointerface have been shown to possess unique properties, such as reduced Auger recombination rates and lower threshold for amplified spontaneous emission [84]. Alloy QDs and graded interface core/shell hetero-NCs can also be directly synthesized and have attracted increasing interest in the last few years, leading to the investigation of several II-VI and IV-VI compositions [viz., Cd(Te,Se), Cd(S,Se), Pb(S,Se), (Cd,Zn)Se, (Cd,Zn)S, (Cd,Zn)(S,Se)] [80–89].

Another effective strategy to impart novel properties (e.g., optical or magnetic) to semiconductor NCs is the intentional introduction of impurities (doping) [90]. Doping of bulk materials is a very well developed field, which underpins most of our present technologies, since the properties of materials for lighting, electronic and optoelectronic applications are largely controlled by dopants. In contrast, the precise doping of NCs is still an underdeveloped field, which is however booming and has delivered great successes and many novel materials in recent years [28, 91–100].
Over the last few years, potential toxicity and environmental impact have become important driving forces in the quest for novel semiconductor NCs and hetero-NCs [32], since the best-developed systems to date are based on Cd- and Pb-chalcogenides. A remarkable degree of control over size, shape and composition has been achieved for these types of NCs [14–16, 19, 21, 26, 28, 61], but widespread deployment into consumer products is severely limited by toxicity concerns. This has motivated an increasing research effort on alternative compositions that are based on less toxic elements, such as copper chalcogenides (e.g., CuInS2) [24, 26, 32], InP [26, 101], and Si [102].

2.3 Nanoscale Surfaces: far from “Superficial”

The most prevalent nanoscale effect is the increase in the surface-to-volume ratio with decreasing size. Surface atoms comprise only a very small fraction of the constituents of bulk solids, and therefore have a negligible contribution to the material properties. In contrast, the fraction of atoms at surfaces and/or (hetero) interfaces is significant at the nanoscale and becomes increasingly larger as the NC dimensions are further reduced. As a result, the contribution of the surface atoms to the properties of the NC becomes increasingly larger, eventually giving rise to completely novel properties. Surface atoms have fewer neighbors, and therefore possess a higher free energy and unsatisfied chemical bonds (the so-called dangling bonds). The increasingly larger surface/volume ratio of NCs will thus render them more reactive and dynamic than bulk crystals, which impacts a number of properties, such as melting temperatures, solubility, plasticity, catalytic activity, crystal structure, and colloidal dispersibility [16, 103].

The NC surface is a dynamic interface between the inorganic semiconductor core and the ligand shell. The interaction between the semiconductor core and the ligands is crucially relevant during the synthesis of colloidal NCs, since it affects both the thermodynamics and kinetics of their nucleation and growth [16]. It is thus largely responsible for the remarkable degree of control achieved over the size, shape, and composition of semiconductor NCs and hetero-NCs [16]. Another important consequence of the large contribution of surface atoms to the properties of NCs is the enhancement of the solid-state diffusion rates. This has made it possible to use nanoscale cation exchange and/or controlled interdiffusion as post-synthetic strategies to tailor the properties of NCs while preserving their size, shape, and heterostructure, by tuning their composition and/or elemental distribution profile [82, 104–122]. These techniques have also been recently used to achieve doping of semiconductor NCs [96, 97, 100].

The larger surface-to-volume ratio of NCs affects the optoelectronic properties. The best-known effect is that unshared atomic orbitals of surface atoms can give rise to localized energy levels within the HOMO–LUMO gap of the NC, which are known as trap states. These states can be detrimental to the PL quantum yield of the NC, if carrier localization into these states is followed by nonradiative exciton relaxation (i.e., energy dissipation as heat by coupling to vibrations) [16]. Radiative recombination between delocalized and trapped carriers may also occur, giving rise to PL that is strongly red-shifted with respect to the band-gap of the NC (the so-
called trap PL). This emission is typically characterized by very broad bandwidths and low quantum efficiencies. It is thus often desirable to eliminate dangling bonds and defects at the surface of semiconductor NCs. This can be achieved by overcoating the NC either with a shell of a different semiconductor (thus forming a hetero-NC, see Sect. 2.2 above) or with suitable ligands that form strong bonds with the surface atoms, thereby shifting the energies of the surface states away from the HOMO–LUMO gap of the NC [16, 123]. Other ligands may in fact generate localized interfacial states or mid-gap states that trap one of the carriers and induce PL quenching (e.g., hole trapping by alkanethiols on CdSe QDs [82, 123, 124]), or directly shift the NC electronic states due to electrostatic or orbital mixing effects [64, 123, 125–127]. The capping ligand shell can be viewed as a self-assembled monolayer (SAM) at the surface of the NC [16, 123]. The internal structure of this SAM can also affect the PL of the NCs, either positively, by fostering surface reconstruction that eliminates trap states [16], or negatively, by imposing disorder to the surface [16, 128].

2.4 Collective Effects in NC Superstructures: When 1 + 1 is Larger Than 2

An attractive feature of colloidal semiconductor NCs and hetero-NCs is that they can be used as solution-processable building blocks for nanostructured thin-films, either by directly depositing the colloidal suspension of NCs or hetero-NCs (the so-called NC inks) onto a substrate and evaporating the solvent [20, 23], or by allowing the NCs or hetero-NCs to self-organize into long-range three- or two-dimensionally ordered superlattices at air–liquid interfaces and subsequently transferring the superstructure to a suitable substrate [129, 130]. Colloidal NCs can also self-assemble into three-dimensionally ordered colloidal superparticles [131]. The geometry and properties of these superstructures can be tailored by the size, shape, composition and surface chemistry of the NC or hetero-NC building blocks [129–138]. In particular, surface ligands have been shown to have a dramatic impact on the directionality of the self-organization process [135, 139–143], leading in some cases to atomically aligned NC superlattices [135, 139, 143]. NC thin-films and superlattices hold promise for a variety of optoelectronic devices, such as light emitting devices, solar cells, photodetectors, and field-effect transistors [129, 130], since they may give rise to a number of novel properties dominated by collective interactions such as energy transfer, charge carrier transfer and migration, and inter-NC electronic coupling.

Another interesting type of superstructure is obtained by attaching colloidal semiconductor NCs (typically CdSe, CdTe, PbSe, or CuInS₂) to nanostructured mesoporous films of wide band gap oxide semiconductors, such as TiO₂ or SnO₂. Depending on the band alignments, fast electron injection from the NC into the mesoporous film will occur, making it possible to use such superstructures as QD-sensitized solar cells, akin to the well-known dye-sensitized Grätzel solar cells [144–147].
3 Excited-State Dynamics in Semiconductor Nanocrystals

In this section, we discuss the excited-state dynamics of semiconductor NCs and hetero-NCs, i.e., the processes that occur in a NC after excitation eventually leading to the emission of light. Following excitation, a NC makes the transition from one level to another until eventually relaxing back to the ground state. This sequence of events involves time scales that span over 15 orders of magnitude, from a few femtoseconds to a few seconds after photoexcitation. The possible relaxation pathways and the balance between their rates determine how efficiently light is emitted and at what wavelength. We will limit our discussion to processes occurring at room temperature, since these are more relevant for potential applications, and will thus neglect exciton fine-structure effects (see Sect. 2.1 above). Excited-state dynamics that are dictated by inter-NC interactions, such as energy migration [35, 130, 148–150] and charge carrier transport [130, 151], or energy transfer between NCs and acceptor molecules (usually referred to as Förster resonance energy transfer, FRET) [31, 35, 152] are also beyond the scope of this review, and we refer the interested reader to prior publications that focus on these aspects.

3.1 Relaxation of Hot-Carrier States: fs to ps Timescales

Directly after photoexcitation a NC is in a high-energy state, where usually both electron and hole occupy levels deep in the conduction and valence band. In other words, they have energy in excess of the band edge, and are usually referred to as “hot carriers”. Typically the excess carrier energy is rapidly lost as heat on a picosecond timescale or faster [153–161] (Fig. 4a). Studies of this cooling process, and attempts to suppress it, have until now mainly focused on Cd- and Pb-based NCs, using photoluminescence spectroscopy and transient absorption spectroscopy. The mechanism of rapid cooling is not precisely known, but thought to involve coupling to vibrations as well as Auger-coupling between electrons and holes [51, 162–165].

There has been interest in making use of hot-carrier energy in NCs, by reducing the cooling efficiency. The possibility of multi-exciton generation (MEG), also called carrier multiplication (CM), has been investigated for several years, most commonly in Pb-chalcogenide NCs [166–172], but also for other NC materials [173–176]. In the process of multi-exciton generation, a hot carrier with excess energy higher than the bandgap can relax to the ground state while generating an additional electron–hole pair (Fig. 4b). This process has been theoretically predicted to happen on a fs to ns timescale [177]. Multi-exciton generation has the potential to increase the efficiency of QD solar cells to above the Shockley-Queisser limit [178], and is therefore of great interest (see Fig. 5). Many studies have reported the possibility of efficient MEG in NCs, but other studies have challenged too optimistic values for efficiency and energy threshold [169, 175, 176]. As a variation to multi-exciton generation in individual QDs, the phenomenon of space-separated quantum cutting has been reported for ensembles of QDs of Si. Here, the hot-carrier
energy in one QD is transferred to a neighboring QD, after which both can emit.

NCs can show direct light emission from hot-exciton states (Fig. 4c). Interband hot-carrier emission is due to recombination of a hot carrier in one band (e.g., an electron in the conduction band), with a carrier in the other band. This emission is blue-shifted with respect to that from the ground-state exciton, and decays on a timescale of picoseconds or faster [73, 172, 181]. In addition, the possibility of intraband hot-carrier emission has recently been demonstrated in Cd-based and Hg-based NCs [182, 183]. In this process, a hot carrier relaxes to a lower energy level in the same band by the emission of an infrared photon. A particular variation of hot-carrier interband emission occurs in hetero-NCs, if charge carrier localization to the equilibrium situation (as according to the band alignment; see Fig. 3) is inhibited. For example, in CdSe/CdS hetero-NCs hole localization from the high-bandgap material CdS to the CdSe core can be suppressed at high excitation power when multiple mutually repulsive valence band holes simultaneously co-exist in the hetero-NC (Coulomb blockade effect). It has been shown that this leads to significant emission from the CdS arms in CdSe/CdS tetrapods [184–186] or from the CdS shell in CdSe/CdS dot-in-bulk NCs [187].

Another way to reduce the loss of hot-carrier energy is to offer charge transfer pathways that compete with cooling (Fig. 4d). To achieve this, a charge transfer time constant of at most a few picoseconds is necessary. Hot-electron transfer on femtosecond timescales has been demonstrated from PbSe QDs to TiO$_2$ [188]. Moreover, transfer from hot-carrier states inside the QD to states on the surface or in the environment has been proposed to contribute to photo-ionization and blinking.

Fig. 4 Schematic representation of possible relaxation pathways for hot-carrier states. a Thermalization by means of electron–hole Auger coupling and/or coupling to vibrations. b Multi-exciton generation, where the hot-carrier energy is converted into an additional electron–hole pair. c Hot-exciton emission. d Ejection of a hot charge carrier to the environment of the NC.
Charge transfer from hot-electron states is of potential use for solar cell applications, where currently thermalization constitutes a major part of the energy conversion losses [191] (see also Fig. 5b). To enable efficient hot-electron transfer, the competing process of cooling must be suppressed. This could be achieved in designed hetero-NCs to decouple the hot electron from the hole in the valence band as well as from ligand vibrations [51].

3.2 Auger Decay of Multi-Carrier States: ps to ns Time Scales

After the carriers have cooled down to the edges of valence and conduction band via the pathways depicted in Fig. 4, the next important relaxation pathway is Auger decay. An Auger process is the transfer of energy from one charge carrier in the NC
to another. This process plays an important role in semiconductor NCs whenever there are three or more charge carriers present, of which at least one electron and one hole. The most common NC states likely to undergo an Auger process are trion states, i.e., charged states with an electron–hole pair in the NC as well as an additional charge carrier in a quantum confined energy level, and the biexciton state, i.e., the state with two electron–hole pairs in the NC. Figure 6 illustrates the most commonly considered Auger processes. These are Auger processes that involve charge carriers in delocalized levels (i.e., the quantum confined orbitals that extend over the entire NC). The final state of an Auger process always has a charge carrier in a highly excited level (situations II in Fig. 5), which then usually undergoes rapid cooling as in Fig. 4a. The net result of an Auger process is therefore energy loss as heat. Trapped charge carriers can be involved in Auger processes in NCs [192–194]. Auger processes involving trapped charge carriers are poorly understood and not further discussed here, but should be investigated further.

For the use of semiconductor NCs as photoluminescent centers, e.g., in LEDs, laser gain material, or biomedical tracers, it is usually desired that optical cycling is as efficient as possible. This means that every photon absorbed should lead to a photon emitted. Therefore, to minimize Auger losses, the NC must be uncharged (i.e., no trion Auger decay; Fig. 6a, b) and the excitation intensity must be sufficiently low to prevent the generation of biexcitons (Fig. 6c). However, under

---

**Fig. 6** Examples of Auger processes in semiconductor NCs in the trion (i.e., charged) or biexciton state.  
**a** In the positive trion state, the recombination energy of an electron–hole pair can be transferred to the additional hole.  
**b** In the negative trion state, the recombination energy of an electron–hole pair can be transferred to the additional electron.  
**c** In the biexciton state, the recombination energy of an electron–hole pair can be transferred either to the additional hole (the “positive trion pathway”; to the left) or to the additional electron (the “negative trion pathway”; to the right) [201]
illumination NCs charge up intermittently and seemingly randomly, leading to a phenomenon known as PL intermittency or “blinking” (see Sect. 3.4 below) [195, 196], while for applications such as lasing high excitation intensities are a necessity [14, 197, 198]. As a result, the possibility of Auger quenching cannot always be avoided. It should be noted that at the highest operating powers in lasers, stimulated emission easily outcompetes Auger recombination, but Auger recombination nevertheless negatively affects the lasing threshold in typical QD lasers.

In recent years, considerable research efforts have been devoted to understanding Auger decay in semiconductor NCs [199–201]. The timescale of Auger processes is typically on the order of 1–100 ps in single component NCs such as PbSe [199], Si [173], PbS [202], CdTe [203], Ge [204], InAs [205], or Pb-perovskites [161]. The Auger timescales in NCs are faster than those in the corresponding bulk material [204]. The difference in Auger rates between bulk (slower) and nanocrystalline (faster) materials, is believed to be due to two reasons. First, all interactions between charge carriers in NCs, including Auger interaction, are enhanced because of their spatial and dielectric confinement (i.e., they are spatially confined in a small volume with high dielectric constant $\varepsilon_1$, which is embedded in a medium with lower dielectric constant $\varepsilon_2$). This increases the Coulomb interaction energy between carriers, which mediates Auger scattering. Second, the conservation rule for translational momentum that suppresses Auger processes in bulk materials is less strict in NCs, because spatial confinement leads to uncertainty in momentum (Heisenberg principle). Figure 7 schematically illustrates the momentum selection rule that governs Auger interaction rates in NCs, using negative trion decay as an example.

The trion Auger dynamics in semiconductor NCs have been investigated under pulsed excitation using photoluminescence measurements. Under strong excitation, NCs charge intermittently (see also Sect. 3.4), which allows one to investigate the

![Fig. 7](image-url) A simple picture of momentum conservation for Auger processes in NCs. Negative trion Auger recombination is depicted in the electronic dispersion diagram of a typical direct-gap semiconductor. a Recombination of an electron–hole pair at the band edge of a bulk semiconductor involves no change in translational momentum (vertical downward arrows). The excess electron can therefore not accept the recombination energy to make an energy and momentum conserving transition. b In a NC, on the other hand, the quantum confined energy levels have no well-defined translational momentum. In other words, the charge carrier wavefunctions contain many spatial frequency components (red and blue shaded areas). The momentum selection rule is therefore not so strict in a NC. Consequently, the Auger process can be much faster, depending on the overlap between charge carrier wavefunctions in momentum space [224, 225]
properties of trions on the single-NC level [72, 206–208]. However, this method is only applicable to NCs where the trion state luminescence can be clearly distinguished from the neutral exciton luminescence. For many types of NCs, the blinking behavior is complicated by the involvement of multiple states [209], and one cannot rely on random charging to investigate charged states. Methods of controlled electrochemical or photochemical charging of luminescent NCs have been developed as an alternative for the studies of luminescence from charged NCs [190, 210, 211]. Not only do these methods offer control over the charge state of QDs, they also allow for statistically significant measurements on entire ensembles [211], while single-NC experiments are necessarily limited to a small number of NCs.

Ensemble transient absorption experiments are most commonly used to study multi-exciton dynamics in NCs [161, 173, 199, 203–205]. Analysis of the fast photoluminescence decay components of NCs under strong excitation can also provide information about multi-exciton decay rates [170, 212]. Both methods rely on a significant multi-exciton population in the NCs, and therefore require strong laser excitation. One must be careful that under such conditions, the interpretation of data can become complicated if NCs charge up or defects are generated [157, 161]. Alternatively, information about biexciton dynamics and quantum efficiencies can be obtained from photon correlation analysis on the single-NC level [213, 214] or the ensemble level [215].

Auger recombination negatively affects the performance of NCs for applications such as light emitting diodes [216], lasers [84], or solar cells under concentrated illumination [217]. As discussed above, Auger processes in NCs are rapid and efficient because of spatial confinement of charge carriers. Therefore, the most obvious way to reduce Auger recombination rates is to increase the size of NCs [199, 201, 204, 211, 218]. However, this may not always be a desired strategy if one wants to make use of quantum confinement effects to tune the electronic properties of NCs. For more subtle control over Auger processes, hetero-NCs have been developed with designed charge carrier confinement potentials [219, 220]. The most commonly studied hetero-NC composition for reduced Auger losses is CdSe/CdS in various shapes and sizes [72, 84, 206, 221–223]. It was first predicted theoretically [224, 225] and later confirmed experimentally [84, 221–223] that an alloyed core–shell interface leads to suppressed Auger recombination. An alloyed hetero-interface creates a smooth confinement potential for charge carriers in which high-momentum components in the wavefunctions are reduced in amplitude (see Fig. 7). Indeed, Auger recombination rates in NCs are strongly dependent on the exact size and shape of the NC [201], leading to wide variations in Auger dynamics within a NC ensemble [226, 227]. Hence, it seems that the careful design of uniform ensembles of hetero-NCs with smooth confinement potentials is the pathway to NCs with reduced Auger losses.

3.3 Radiative Decay in Semiconductor Nanocrystals: ns to μs Time Scales

The most studied electron–hole recombination channel in semiconductor NCs is spontaneous radiative decay. Not only do many experimental methods rely on the
detection of photons emitted in a radiative decay pathway, but radiative decay of excited NC states is also often the desired pathway for applications. Several types of spontaneous radiative decay are possible in semiconductor NCs, depending on the composition. For most NC compositions under moderate excitation, radiative decay is predominantly due to recombination of two delocalized charge carriers (electron and hole) in the lowest-energy quantum confined states of the two respective bands. The electron and hole wavefunctions, and hence the characteristics of the emission, are determined by the composition, size and geometry of the NC, as illustrated in Figs. 1, 2 and 3. With the development of NCs and hetero-NCs of a wide variety of sizes, shapes, and compositions, precise control over spontaneous emission from the lowest-energy exciton has been achieved. Radiative recombination of a delocalized carrier with a trapped (i.e., localized) carrier may also occur (trap PL), but is usually characterized by low quantum efficiencies, since carrier trapping favors non-radiative decay pathways by decreasing the electron–hole wave function overlap while increasing the coupling between the localized carrier and its immediate vicinity. In the case of NCs doped with luminescent ions (e.g., ZnSe:Mn$^{2+}$ [228] or LaPO$_4$:Tb$^{3+}$ [229]), radiative recombination occurs primarily at the dopant.

In this section, we discuss what determines the photoluminescence quantum efficiency of NCs and the rate of radiative decay. Auger quenching diminishes the quantum efficiency of the emission when a NC is charged or when multiple electron–hole pairs are present (see Sect. 3.2 above). However, uncharged NCs under weak illumination may find alternative non-radiative decay pathways, which lower the quantum efficiency. Below, spontaneous radiative decay in semiconductor NCs is first discussed, followed by non-radiative decay pathways that can lead to a photoluminescence quantum efficiency below unity.

The rate of radiative decay of a delocalized electron–hole pair in a semiconductor NC can be estimated as [36, 230]

$$\gamma_{\text{rad}} = C\lambda^{-1}\rho K.$$  

(1)

Here $C$ is a pre-factor that depends on the type of semiconductor, $\lambda$ is the emission wavelength, and $\rho$ is the density of optical states experienced by the exciton (see below for further explanation). $K$ is the electron–hole overlap integral squared.

Figure 8a shows typical values for the radiative lifetime (the inverse of the radiative decay rate) in common and emerging types of zero-dimensional NCs (i.e., QDs): lead chalcogenides (PbE; E = S, Se) [231, 232], copper indium chalcogenides [233, 234] (CuInE$_2$; E = S, Se), cadmium chalcogenides (CdE, E = Se, Te) [58, 235], indium phosphide (InP) [48], silicon (Si) [236], and cesium lead halides (CsPbX$_3$) [161, 237]. A rough trend is visible that radiative lifetimes are longer for longer emission wavelengths. Indeed, Eq. (1) shows that, for a given QD material, the radiative lifetime (which is the inverse of $\gamma_{\text{rad}}$) should scale linearly with emission wavelength. Nevertheless, the various QD materials deviate from a general linear dependence between radiative lifetime and emission wavelength (dashed line). These deviations are due to variations in the pre-factor $C$ and the density of optical states $\rho$. 

Springer
The pre-factor $C$ is a material-specific constant that accounts for the electronic properties of the QD material. In PbE, CdE, InP, and CsPbX$_3$, where the radiative decay is due to recombination of delocalized electron–hole pair (i.e., the lowest energy exciton), $C$ depends for example on how strongly light couples valence and conduction band states, and also the exciton fine-structure [36].

The density of optical states $\rho$ is a factor of potentially large influence on the radiative decay. It depends on the refractive index of the QD, the shape of the QD, and the polarization of the emission, as well as on the photonic environment of the QD. For example, photonic crystals [238] or plasmonic structures [239] can enhance or suppress radiative decay of QDs. For QDs dispersed in an organic medium, as they are commonly prepared and analyzed, the density of optical states $\rho$ is

---

**Fig. 8** The rate of radiative decay in colloidal quantum dots. **a** Typical radiative lifetimes and emission wavelengths obtainable for some common and emerging QDs: lead chalcogenides (PbE with E = S, Se; brown) [231, 232], copper indium chalcogenides (CuInE$_2$ with E = S, Se; yellow) [233, 234], cadmium chalcogenides (CdE with E = Se, Te; red) [58, 235], indium phosphide (InP; green) [48], silicon (Si; gray) [236], and cesium lead halides (CsPbX$_3$ with X = Cl, Br, I; blue) [161, 237]. The differently colored clouds indicate the approximate range of combinations for radiative lifetime and emission wavelength that can be found in these materials. The **dashed line** denotes a linear dependence between radiative lifetime and emission wavelength (see text for details). **b** The density of optical states $\rho$ (Eq. 2) for QDs of various semiconductor materials dispersed in a medium with refractive index $n = 1.5$, such as toluene or poly(methyl methacrylate), normalized to the density of optical states in vacuum. The refractive index data were taken from Ref. [306] for CsPbBr$_3$ (blue line), from http://www.filmetrics.com for CdTe (red), InP (green), and PbS (brown), and from Ref. [307] for CuGaS$_2$ (yellow; as a close analogue of CuInE$_2$ materials)
determined by the refractive index of the medium $n$ and the refractive index contrast with the QD material itself $n_{\text{QD}}$:

$$\rho = n \left| \frac{3n^2}{2n^2 + n_{\text{QD}}^2} \right|^2.$$  \hspace{1cm} (2)

Indeed, the radiative decay rate of excitons in core–shell QDs [240] and of luminescent doped ions in NCs [226] depends on the solvent refractive index as described by Eq. (2). The factor $\left| \frac{3n^2}{2n^2 + n_{\text{QD}}^2} \right|^2$ in Eq. (2) is also known as the local-field factor, and describes the effect of the refractive index contrast between the QD and the surrounding medium. Figure 8b shows that the values of the local-field factor range over one order of magnitude for different QD materials.

The electron–hole overlap integral squared [230] can be expressed as

$$K = \left| \int \psi_e(r) \psi_h(r) dr \right|^2 \hspace{1cm} (3)$$

where $\psi_e(r)$ and $\psi_h(r)$ are the electron and hole wavefunctions. The factor $K$ is commonly used to reduce the radiative decay rate of excitons in NCs, by making type-I$\frac{1}{2}$ or type-II hetero-NCs in which the electron and hole are spatially separated [241] (see Sect. 2.2; Fig. 3 above). Reduced rates of spontaneous radiative decay can be useful for applications such as in lasers or photodetectors, where spontaneous emission is not the desired decay pathway for charge carriers. The mechanism of radiative recombination in some QD materials such as Si or CuInE$_2$ is believed to involve at least one localized charge carrier [24, 32, 236, 242], and therefore cannot be expected to follow the trend predicted by Eq. (1). It is not yet clear how hetero-NCs with engineered electron and hole wavefunctions can be used to control radiative decay rates in QDs of materials such as Si or CuInE$_2$ [117].

The quantum efficiency $\eta$ of NC emission is determined by the competition between the radiative decay rate $\gamma_{\text{rad}}$ and all the possible non-radiative decay pathways with a combined rate $\gamma_{\text{nr}}$:

$$\eta = \frac{\gamma_{\text{rad}}}{\gamma_{\text{rad}} + \gamma_{\text{nr}}} \hspace{1cm} (4)$$

As discussed in Sect. 2.3 above, imperfections in the NC such as crystal defects or unsaturated chemical bonds on the surface have been identified as an important factor determining the quantum efficiency [123]. They provide “trap states” for charge carriers, i.e., energy levels within the bandgap where the charge carrier is spatially localized. Indeed, the quantum efficiency of NC emission improves when the NC surface is covered with a protective shell of high-bandgap material [13, 15, 16, 61, 74, 243–246], or when ligands saturate chemical bonds on the surface [16, 123, 247–251].

Generally, the quantum efficiency of different NCs within a single synthesis batch varies strongly. Some NCs have a high quantum efficiency (near 100 %), while others have a quantum efficiency near 0 %. The two subpopulations in a NC
ensemble are also known as the “bright fraction” and the “dark fraction” [251–256] (see Fig. 9). This means that within a single batch, some NCs have (almost) no non-radiative decay pathways, while in other NCs non-radiative decay is very likely. The dark fraction hardly contributes to the observed photoluminescence and, as a result, the photoluminescence decay measurements will reflect the dynamics of the bright NCs. This makes it possible that the photoluminescence decay curve for NC samples with ensemble quantum efficiencies well below 100 % are nearly single-exponential, with a time constant equal to the radiative lifetime of the NCs. High-quality NC batches are thus brighter not necessarily because non-radiative decay is

**Fig. 9** The fraction of dark quantum dots in a sample. **a** With increasing quality of a QD sample, the photoluminescence efficiency improves (blue line). This happens not only because the brightness of each individual QD increases, but mainly because the number of completely dark QDs in the sample decreases (red line). **b** A low-quality sample contains many dark QDs and only a few bright ones. **c** A high-quality sample contains fewer dark QDs. **d** The surface of a dark CdSe QD: Cd (100) surface with a single Cd(oleate)$_2$ ligand attached, leaving many unsaturated surface atoms, which are believed to act as charge carrier traps enabling non-radiative recombination. **e** The sample quality improves when the QD surfaces are covered by ligands [251]. However, the maximum ligand coverage set by steric hindrance (≈ 3 nm$^{-2}$ for oleic acid) is not sufficient to saturate all surface atoms (≈ 6 nm$^{-2}$ for CdSe {100} or CdSe {111}) [123]. There are still unsaturated surface atoms, as highlighted with a question mark. Panels **d** and **e** were adapted from Ref. [123]
suppressed in each *individual* NC, but rather because the fraction of completely dark NCs in the ensemble is smaller. This is schematically depicted in Fig. 9a–c.

Despite over 20 years of research, the mechanisms of photoluminescence quenching by charge carrier trapping are still poorly understood. The simplest picture is that quenching can be suppressed by saturating chemical bonds of the surface atoms [15, 16, 61, 243–249, 251]. This can be achieved by overcoating the NC either by a shell of another semiconductor or by a ligand layer. For example, the photoluminescence quantum efficiencies of CdSe QDs can be increased to values as high as 85% either by overcoating with CdS shells or by capping with primary alkylamines such as hexadecylamine. Computational studies have shown that linear chain alkylamines can form densely packed monolayers at the surface of CdSe NCs, saturating all the available surface Cd atoms [257]. Nevertheless, bright NCs can exist in a sample with low photoluminescence quantum efficiencies, low average surface quality and low average surface coverage by ligands (Fig. 9b, d). Many commonly used surface ligands, such as oleic acid, are bulky and can only saturate half of the available surface atoms because of steric hindrance [123]. This means that even the brightest NCs in an ensemble of oleic acid capped NCs have imperfect saturation of surface atoms. Clearly, additional factors affect the electronic structure of the NC surface, and charge carrier trapping. For example, trap states can be eliminated by surface relaxation and/or reconstruction in such a way that the dangling orbitals of neighboring cations and anions partially overlap, leading to a redistribution of electronic density that makes the surface auto-compensated (a process known as self-passivation or “self-healing”) [258]. Surface- and global reconstruction has been observed for NCs of several compositions (e.g., CdSe, ZnSe) [259, 260], and shown to be affected by the nature and structure of the capping ligand monolayer [128, 261]. Nevertheless, it is currently unknown how the atomic structures, including surface ligands, of bright and dark NCs differ [262]. Some recent successes in correlated optical and electron microscopy have been reported, that can lead to more insight into the microscopic nature of quenching [222, 263, 264]. For example, a lower (time-average) brightness of NCs has been connected to stacking faults in the crystal structure or imperfections in surface coverage by a high-bandgap semiconductor shell [263]. The very dynamic nature of the NC surface and the strong interplay between capping ligands and the inorganic core [16] are also likely important factors determining charge carrier trapping. Unfortunately, the organic surface ligands are invisible in electron microscopy and can currently be investigated only on the ensemble level using infrared absorption [249, 265, 266], nuclear magnetic resonance spectroscopy [249, 267, 268], or neutron scattering [269, 270].

### 3.4 Blinking Dynamics on ms Timescales and Slower

Interestingly, the timescales relevant to the optical properties of semiconductor NCs extend to much longer than the radiative lifetime of the exciton. In this section, we do not discuss irreversible bleaching of NC luminescence due to, for example, oxidation [271], but only reversible physical phenomena encountered in NCs on timescales beyond the exciton radiative lifetime. These include not only
photoluminescence intermittency (blinking), but also photodimming and photo-brightening, spectral diffusion, and delayed emission. We will first give a brief overview of experimental studies, and then discuss the microscopic nature of blinking and related processes, which is still largely unknown.

In 1996, Nirmal et al. [195] observed that the luminescence from NCs turns on and off intermittently on time scales from milliseconds up to many seconds (see Fig. 10a). This phenomenon, commonly known as “blinking”, becomes apparent in studies on individual NCs, but is hidden in ensemble measurements on many NCs simultaneously. Nevertheless, blinking does have an adverse effect on the properties of NC ensembles, because there is always a fraction of NCs in the non-emissive state. Interestingly, because of the peculiar statistics of blinking, the fraction of non-emissive NCs can change in time under continued excitation, leading to reversible photodimming or photobrightening over time scales of many seconds (Fig. 10b) [272–274].

Blinking is observed in many different types of NCs, including QDs of CdSe [195], InP [275], CdTe [276], PbS [277], InAs [278], Pb-perovskites [279], as well as various hetero-NCs, and even in very different emitters such as organic dye molecules [280]. Methods to reduce blinking of NCs (i.e., to make random switches to a dark state less frequent) include surface protection using organic ligands [281] or an inorganic shell material [195, 282, 283], and also plasmonic enhancement of radiative decay [284]. This indicates that blinking involves slow changes on the surface of the QD that introduce non-radiative decay pathways. These can be either geometrical changes induced by ligand adsorption and desorption, or charge carrier trapping (for more discussion, see below).

A second phenomenon fundamental to semiconductor NCs but only observable in single-emitter measurements, is spectral diffusion (Fig. 10c). This entails that over time the emission spectrum of a NC shifts or jumps back and forth over the range of a few nanometers. Temporal variations in the peak emission wavelength are accompanied by, and correlated with, variations in the emission line width [285–288]. Most of the experiments into spectral diffusion were conducted at cryogenic temperatures [289–292], but the process occurs at room temperature, too [285–288]. A direct link between spectral diffusion and blinking was proposed, based on correlations between blinking events (on–off switches) and spectral shifts [290, 293]. At room temperature, typical time scales for spectral diffusion are milliseconds to seconds, but not shorter [294].

The statistics of blinking are peculiar. Typical duration distributions of bright and dark periods in an emission trace (as in Fig. 10a) are depicted in Fig. 10d. The durations are power-law distributed (with an exponential cut-off for the bright periods at long time scales) [295]. This means that a bright or dark period is most likely short (only a few milliseconds), but much longer periods of many seconds occur as well. The range of durations is much wider than it would be in case of exponential statistics. The power-law exponents are around 1.5 for most NCs, and independent or nearly independent of temperature [276], excitation intensity [276], and nature of the excitation laser (continuous wave or pulsed) [296]. Interestingly, the band-edge emission of semiconductor NCs contains a slow “delayed” component with power-law statistics, that extends over time scales from
nanoseconds up to (at least) milliseconds [240, 256, 297] (Fig. 10e). This very slow emission is not trap emission due to recombination of a trapped and a delocalized charge carrier (see Sect. 2.3 above), because the emission wavelength is (nearly) identical to the band-edge emission. Instead, this emission component has been ascribed to reversible charge carrier trapping and detrapping, followed by emission [298]. Based on the similar statistics, a close relation between blinking and delayed emission has been suggested [240, 297].
The microscopic nature of blinking is, after 20 years, still under debate. Models must contain at least two ingredients: they must explain what makes a NC dark in the off state, and what causes the characteristic power-law statistics. Early on, Efros and Rosen [196] proposed the charging–discharging model, where the NC can become charged by the ejection of a photogenerated charge carrier. A neutral NC would correspond to the on state, while a charged NC is in the off state, where photoluminescence is quenched by Auger recombination (see Sect. 3.2). The initial model [196] proposed that charging could be due to Auger ejection of a charge carrier following the generation of a biexciton, but this would result in exponential statistics. Several adaptations for the charging–discharging model have been developed to account for the power-law statistics of blinking. These adapted models assume that the rate of charging and/or discharging varies in time, because of Coulomb blockade [299] or fluctuations in the geometry and surface structure of the NC [300], or because tunneling barriers for charge carrier trapping vary slowly in height and width [295].

The picture of Auger quenching in the off state has later been challenged, based on comparisons between the quenching rate of the biexciton state (due to Auger processes) and the off state [301]. However, such comparison assumes that Auger quenching in the off state is only due to the remaining charge carrier in the NC, while the ejected charge carrier does not play a role. Taking this role into account [193], may explain the discrepancies between biexciton and off state quenching. As an alternative to charging–discharging model, the multiple recombination center model was proposed in which structural changes in the NC geometry open and close pathways for trapping and non-radiative decay of charge carriers [302, 303]. These models also reproduce the power-law statistics of blinking. However, rapid non-radiative recombination is inconsistent with other experimental data, such as power-law delayed emission [240].

All existing models for blinking have one important weakness: they provide a mathematical description for blinking, but they lack a detailed microscopic (chemical) picture. In fact, it is surprising how little is established about blinking after 20 years of research, other than the statistics. A microscopic picture of blinking may eventually emerge from experiments combined with atomistic quantum mechanical calculations [264, 304] or from very challenging studies of correlated optical and time-resolved electron microscopy [305].

4 Summary and Outlook

The last three decades have witnessed a remarkable development in the colloidal synthesis of composition-, size-, and shape-controlled semiconductor NCs and hetero-NCs, allowing researchers to make materials with tailored physical–chemical and optoelectronic properties by exploiting nanoscale phenomena, such as quantum confinement and surface effects. These effects, and their impact on the properties of semiconductor NCs and hetero-NCs, were discussed in detail in Sect. 2.

The availability of high-quality colloidal nanomaterials has in turn lead to great advances in the fundamental understanding of their properties. In this review, we
focused on the excited-state dynamics in these nanomaterials, covering the whole range of relaxation processes spanning from the fs to the ms time scales: hot carrier relaxation (fs to ps), Auger decay of multi-carrier states (ps to ns), radiative decay (ns to μs), and photoluminescence intermittency (blinking), spectral diffusion, and delayed emission, which take place on time scales longer than ms. It is clear that the scientific community has a reasonably thorough understanding of many of the physical processes involved in the exciton formation and relaxation in semiconductor NCs and hetero-NCs, but there are still many poorly understood aspects and several knowledge gaps. As a result, a comprehensive theoretical framework capable of fully describing the exciton dynamics in semiconductor NCs and hetero-NCs has yet to emerge.

A particularly critical challenge is the understanding of the processes taking place at time scales longer than the radiative lifetime of the exciton, and the development of a detailed microscopic model that can relate these processes to chemical and structural transformations of the NC and/or its immediate vicinity. The understanding of the mechanisms underlying carrier trapping and photoluminescence quenching, and the role of capping ligands therein, is still fragmentary and merit a systematic and comprehensive investigation. Progress in this direction has been hampered by the lack of suitable tools, but many new techniques have appeared in recent years, and it is likely that further developments will make these issues amenable to experimental and computational investigation in the near future.

Another current limitation is that the large majority of the studies of the exciton dynamics in semiconductor NCs and hetero-NCs have been carried out on the prototypical case of CdSe and other Cd- and Pb-based compositions, while studies on emerging compositions such as InP, Cu chalcogenides and Si have been scarce. As a result, the latter class of nanomaterials is as yet poorly understood, despite their great potential as sustainable and less toxic alternatives to the conventional Cd- and Pb-based NCs and hetero-NCs. Recent advances in the synthesis of colloidal nanocrystals of these alternative compositions, and the growing interest that they have been attracting, will certainly lead to major efforts to close the gap in the understanding of their properties.

Acknowledgments FTR acknowledges financing by the Netherlands Organisation for Scientific Research (NWO): Rubicon Grant 680-50-1509.

Open Access This article is distributed under the terms of the Creative Commons Attribution 4.0 International License (http://creativecommons.org/licenses/by/4.0/), which permits unrestricted use, distribution, and reproduction in any medium, provided you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made.

References

1. Ekimov AI, Onuschchenko AA (1981) JETP Lett 34:345–348
2. Brus LE (1983) J. Chem. Phys. 79:5566–5571
3. Ekimov AI, Onuschchenko AA (1984) JETP Lett 40:1136–1138
4. Weller H, Koch U, Gutierrez M, Henglein A (1984) Ber. Bunsen. Phys. Chem. 88:649–656
5. Ekimov AI, Efros AL, Onushchenko AA (1985) Solid State Commun 56:921–924
6. Brus LE (1986) J. Chem. Phys. 90:2555–2560
7. Aliivisatos AP, Harris AL, Levinos NJ, Steigerwald ML, Brus LE (1988) J. Chem. Phys. 89:4001–4011
8. Bawendi MG, Kortan AR, Steigerwald ML, Brus LE (1989) J. Chem. Phys. 91:7282–7290
9. Henglein A (1989) Chem Rev 89:1861–1873
10. Weller H (1993) Angew. Chem. Int. Ed. Engl. 32:41–53
11. Murray CB, Norris DJ, Bawendi MG (1993) J Am Chem Soc 115:8706–8715
12. Katari JEB, Colvin VL, Aliivisatos AP (1994) J Phys Chem 98:4109–4117
13. Hines MA, Guyot-Sionnest P (1996) J Phys Chem 100:468–471
14. Klimov VI, Ivanov SA, Nanda J, Achermann M, Bezel I, McGuire JA, Piryatinski A (2007) Nature 447:441–446
15. Talapin DV, Lee J, Kovalenko MV, Shevchenko EV (2010) Chem Rev 110:389–458
16. Donega CdM (2011) Chem Soc Rev 40:1512–1546
17. Zhao Y, Burda C (2012) Energy Environ Sci 5:5564–5576
18. Freeman R, Willner I (2012) Chem Soc Rev 41:4067–4085
19. Shirasaki Y, Supran G, Bawendi M, Bulović V (2012) Nat Photonics 7:13–23
20. Bucherl CN, Oleson KR, Hillhouse HW (2013) Curr. Opin. Chem. Eng. 2:168–177
21. Chen O, Zhao J, Chauhan VP, Cui J, Wong C, Harris DK, Wei H, Han HS, Fukumura D, Jain RK, Bawendi MG (2013) Nat Mater 12:445–451
22. Krummer Z, Pera SI, van Dijk-Moes RJ, Zhao Y, de Brouwer AF, Groeneveld E, van Sark WGHJM, Schropp REI, de Mello Donega C (2013) Sol Energy Mater Sol Cells 111:57–65
23. Stolle CJ, Harvey TB, Korgel BA (2013) Curr. Opin. Chem. Eng. 2:160–167
24. Kolny-Olesiak J, Weller H (2013) ACS Appl. Mater. Interfaces 5:12221–12237
25. Banin U, Ben-Shahar Y, Vinokurov K (2014) Chem Mater 26:97–110
26. Grim JQ, Manna L, Moreels I (2015) Chem Soc Rev 44:5979–5914
27. Knowles KE, Alivisatos AP, Lu, Alzate DG, McDowall S, Gamelin DR (2015) Chem Commun 51:9129
28. Kovalenko MV, Manna L, Cabot A, Hens Z, Talapin DV, Kagan CR, Klimov VI, Rogach AL, Reiss P, Milliron DJ et al (2015) ACS Nano 9:1012–1057
29. Meinardi F, McDaniel H, Carulli F, Colombo A, Velizhanin KA, Makarov NS, Simonutti R, Klimov VI, Brovelli S (2015) Nat Nanotech 10:878
30. Silvi S, Credi A (2015) Chem Soc Rev 44:4275–4289
31. Wegner KD, Hildebrandt N (2015) Chem Soc Rev 44:4792–4834
32. van der Stamm W, Berends AC, Donega CdM (2016) ChemPhysChem 17:559–581
33. Aliivisatos AP (1996) J Phys Chem 100:13226–13239
34. Groeneveld E, Delerue C, Allan G, Niquet Y, de Mello Donega C (2012) J Phys Chem C 116:23160–23167
35. Koole R, Groeneveld E, Vanmaekelbergh D, Meijerink A, Donega CdM (2014) In: Donega CdM (ed) Nanoparticles: workhorses of Nanoscience, Ch. 2. Springer, Heidelberg
36. Efros AL, Rosen M, Kuno M, Nirmal M, Norris DJ, Bawendi MG (1996) Phys Rev B 54:4833–4856
37. Franceschetti A, Fu H, Wang LW, Zunger A (1999) Phys Rev B 60:1819–1829
38. Leung K, Pokrant S, Whaley KB (1998) Phys Rev B 57:12291–12301
39. Crooker SA, Barrick T, Hollingsworth JA, Klimov VI (2003) Appl Phys Lett 82:2793–2795
40. Labeau O, Tamarat P, Lounis B (2003) Phys Rev Lett 90:257404
41. Donega CdM, Bode M, Meijerink A (2006) Phys Rev B 74:085320
42. Wang H, de Mello Donega C, Meijerink A, Glasbeek M (2006) J Phys Chem B 110:733–737
43. Zhao Q, Graf PA, Jones WB, Franceschetti A, Li J (2007) Wang, Kim K. Nano Lett 7:3274–3280
44. Oron D, Aharoni A, de Mello Donega C, van Rijssel J, Meijerink A, Banin U (2009) Phys Rev Lett 102:177402
45. Schaller RD, Crooker SA, Bussian DA, Pietryga JM, Joo J, Klimov VI (2010) Phys Rev Lett 105:067403
46. Eilers J, van Hest J, Meijerink A, Donega CdM (2014) J Phys Chem C 118:23313–23319
47. Siebers B, Biadala L, Yakovlev DR, Rodina AV, Aubert T, Hens Z, Bayer M (2015) Phys Rev B 91:155304
48. Biadala L et al (2016) ACS Nano 10:3356–3364
49. Stroscio MA, Dutta M (2001) Phonons in Nanostructures. Cambridge University Press, New York
50. Klimov VI (2007) Annu Rev Phys Chem 58:635–673
51. Pandey A, Guyot-Sionnest P (2008) Science 322:929–932
52. Kambhampati P (2011) Acc Chem Res 44:1–13
53. Hannah DC, Dunn NJ, Ithurria S, Talapin DV, Chen LX, Pelton M, Schatz GC, Schaller RD (2011) Phys Rev Lett 107:177403
54. Henderson B, Imbusch GF (1989) Optical Spectroscopy of Inorganic Solids. Oxford University Press, Oxford
55. Cui J et al (2016) Nano Lett 16:289–296
56. Norris D, Efros A, Rosen M, Bawendi M (1996) Phys Rev B 53:16347–16354
57. Granados del Águila A, Jha B, Pietra F, Groeneveld E, de Mello Donega C, Maan JC, Vanmaekelbergh D, Christianen PCM (2014) ACS Nano 8:5921–5931
58. Donega CdM, Koole R (2009) J Phys Chem C 113:6511–6520
59. Moreels I, Lambert K, De Muynck D, Vanhaecke F, Poelman D, Martins JC, Allan G, Hens Z (2007) Chem Mater 19:6101
60. Ivanov SA, Piryatinski A, Nanda J, Tretiak S, Zavadil KR, Wallace WO, Werder D, Klimov VI (2007) J Am Chem Soc 129:11708–11719
61. Reiss P, Protière M, Li L (2009) Small 5:154–168
62. Halpert JE, Porter VJ, Zimmer JP, Bawendi MG (2006) J Am Chem Soc 128:12590
63. Zhong H, Scholes GD (2009) J Am Chem Soc 131:9170
64. Donega CdM (2010) Phys. Rev. B 81:165303
65. Pandey A, Guyot-Sionnest P (2007) J. Chem. Phys. 127:104710
66. Oron D, Kazeres M, Banin U (2007) Phys. Rev. B 75:035330
67. Wijnen EJP, Blokland JH, Chin PTK, Christianen PCM, Maan JC (2008) Phys. Rev. B 78:235318
68. Biadala L, Louyer Y, Tamarat Ph, Lounis B (2009) J Phys Chem C 113:16240–16250
69. García-Santamaría F, Chen Y, Vela J, Schaller RD, Hollingsworth JA, Klimov VI (2009) Nano Lett 9:3482–3488
70. Dennis AM, Mangum BD, Piryatinski A, Park YS, Hannah DC, Casson JL et al (2012) Nano Lett 12:5545–5551
71. Rabouw FT et al (2013) Nano Lett 13:4884–4892
72. Diroll BT, Turk ME, Gogotsi N, Murray CB, Kikkawa JM (2016) Chem Phys Chem 17:759–765
73. Abel KA, Qiao H, Young JF, Van Veggel FCJM (2010) J Phys Chem Lett 1:2334–2338
74. Zhao Y, Riemersma C, Pietra F, Koole R, Donega CdM, Meijerink A (2012) ACS Nano 6:9058–9067
75. Brovelli S et al (2011) Nature Commun 2:280
76. Raino G, Stofeर T, Moreels I, Gomes R, Hens Z, Mahrt RF (2012) ACS Nano 6:1979–1987
77. Biadala L, Siebers B, Gomes R, Hens Z, Yakovlev DR, Bayer M (2014) J Phys Chem C 118:22309–22316
78. Granados del Águila A, Groeneveld E, Maan JC, de Mello Donega de, Christianen PCM (2016) ACS Nano 10:4102–4110
79. Zhong X, Han M, Dong Z, White TJ, Knoll W (2003) J Am Chem Soc 125:8589–8594
80. Panda SK, Hickey SG, Waurisch C, Eychmuller A (2011) J Mater Chem 21:11550–11555
81. Groeneveld E, Witteman L, Lefferts M, Ke X, Bals S, van Tendeloo G, Donega CdM (2013) ACS Nano 7:7913–7930
82. Vaxenburg R, Lifshitz E (2012) Phys Rev B 85:075304
83. Petrov DV, Santos BS, Pereira GAL, de Mello Donega C (2002) J Phys Chem B 106:5325–5334
84. Bailey RE, Nie S (2003) J Am Chem Soc 125:7100–7106
85. Maikov GI, Vaxenburg R, Sashchiuk A, Lifshitz E (2010) ACS Nano 4:6547–6556
86. Regulacio MD, Han M (2010) Acc Chem Res 43:621–630
87. Zhang J, Yang Q, Cao H et al (2016) Chem Mater 28:618–625
88. Buonsanti R, Milliron DJ (2013) Chem Mater 25:1305–1317
89. Orlovskii SB, Schmidt J, Groenen EJJ, Baranov PG, de Mello Donega C, Meijerink A (2005) Phys Rev Lett 94:097602
90. Norris DJ, Efros AL, Erwin SC (2008) Science 319:1776–1779
91. Beaucal R, Schneider L, Archer PL, Bacher G, Gamelin DR (2009) Science 325:973–976
92. Bussian DA, Crooker SA, Yin M, Brynda M, Efros AL, Klimov VI (2009) Nature Mater 8:35–40
93. Chen D, Viswanatha R, Ong GL, Xie R, Balasubramaninan M, Peng X (2009) J Am Chem Soc 131:9333–9339
96. Eilers J, Groeneveld E, Donega CDM, Meijerink A (2012) J Phys Chem Lett 3:1663–1667
97. Sahu A, Kang MS, Kompch A, Notthoff C, Wills AW, Deng D, Winterer M, Frisbie CD, Norris DJ (2012) Nano Lett 12:2587–2594
98. Zhao Y, Rabouw FT, van Puffelen T, van Walree CA, Gamelin DR, de Mello Donega C, Meijerink A (2014) J Am Chem Soc 136:16533–16543
99. Schimpf AM, Knowles KE, Carroll GM, Gamelin DR (2015) Acc Chem Res 48:1929–1937
100. Barrows CJ, Chakraborty P, Kornowske LM, Gamelin DR (2016) ACS Nano 10:910–918
101. Ott FD, Spiegel LL, Erwin SC (2014) Phys Rev Lett 133:156803
102. van der Stam W, Bladt E, Rabouw FT, Bals S, Donega CDM (2015) ACS Nano 9:11430–11438
103. van der Stam W, Berends AC, Rabouw FT, Willhammar T, Ke X, Meeldijk JD, Bals S, Donega CDM (2015) Chem Mater 27:621–628
104. de Trizio L, Manna L (2016) Chem Rev. doi:10.1021/acs.chemrev.5b00739
105. Boneschanscher MP, Evers WH, Byelor DV, Petukhov AV, Bals S, Vanmaekelbergh D (2012) Nano Lett 12:5515–5523
106. Ye X, Chen J, Engel M, Millan JA, Li W, Xing G, Collins JE, Kagan CR, Li J, Glotzer SC, Murray CB (2013) Nat Chem 5:466–473
136. van der Stam W, Gantapara AP, Akkerman QA, Soligno G, Meeldijk JD, van Roij R, Dijkstra M, de Mello Donega C (2014) Nano Lett 14:1032–1037
137. van Anders G, Klotsa D, Karas AS, Dodd PM, Glotzer SC (2015) ACS Nano 9:9542–9553
138. Vanmaekelbergh D, van Vugt LK, Bakker HE, Rabouw FT, De Nijs B, van Dijk-Moes RJA, van Huis MA, Baesjou PJ, van Blaaderen A (2015) ACS Nano 9:3942–3950
139. Choi JJ et al (2011) J Am Chem Soc 133:3131–3138
140. Nagaoka Y, Chen O, Wang Z, Cao YC (2012) J Am Chem Soc 134:2868–2871
141. Wang Z, Schliehe C, Bian K, Dale D, Bassett WA, Hanrath T, Klinke C, Weller H (2013) Nano Lett 13:1303–1311
142. Goodfellow BW, Yu Y, Bosoy CA, Smilgies DM, Korgel BA (2015) J Phys Chem Lett 6:2406–2416
143. van der Stam W, Rabouw FT, Vonk SJW, Geuchies JJ, Ligthart H, Petukhov AV, Donega CDM (2016) Nano Lett 16:2608–2614
144. Pijpers JIH, Koole R, Evers WH, Houtepen AJ, Böhme S, Donega CDM, Vanmaekelbergh D, Bonn M (2010) J Phys Chem C 114:18866–18873
145. McDaniel H, Fuke N, Makarov NS, Pietryga JM, Klimov VI (2013) Nat Commun 4:2887
146. Jara DH, Yoon SJ, Stamplecoskie KG, Kamat PV (2014) Chem Mater 26:7221–7228
147. Pan Z, Mora-Serro I, Shen Q, Zhang H, Li Y, Zhao K, Wang J, Zhong X, Bisquert J (2014) J Am Chem Soc 136:9203–9210
148. Konstantatos G, Huang C, Levine L, Lu Z, Sargent EH (2005) Adv Func Mater 15:1865–1869
149. Xu F et al (2011) Nano Nano 5:9950–9957
150. Shabaev A, Efros AL, Efros AL (2013) Nano Lett 13:5454–5461
151. Chou KF, Dennis AM (2015) Sensors 15:13288–13325
152. Rabouw FT et al (2015) ACS Nano 9:10366–10376
153. Krahne R et al (2011) ACS Nano 5:9950–9957
154. Hendry E et al (2006) Phys Rev Lett 96:057408
155. Kilina SV, Kilin DS, Prezhdo OV (2009) ACS Nano 3:93–99
156. Rabouw FT et al (2015) ACS Nano 9:10366–10376
157. Schaller RD, Klimov VI (2004) Phys Rev Lett 92:186601
158. Ellingson RJ et al (2005) Nano Lett 5:865–871
159. Trinh MT et al (2008) Nano Lett 8:1713–1718
160. Pijpers JIH et al (2009) Nat Phys 5:814–818
161. Sandberg RL et al (2012) ACS Nano 6:9532–9540
162. Midgett AG et al (2013) Nano Lett 13:3078–3085
163. Cirloganu CM et al (2014) Nat Commun 5:4148
164. Beard MC et al (2007) Nano Lett 7:2506–2512
165. Schaller RD, Pietryga JM, Klimov VI (2007) Nano Lett 7:3469–3476
166. Nair G, Mawardi MG (2007) Phys Rev B 76:081304R
167. Ben-Lulu M, Mocatta D, Bonn M, Banin U, Ruhman S (2008) Nano Lett 8:1207–1211
168. Allan G, Delerue C (2006) Phys Rev B 73:205423
169. Timmerman D, Izeddin I, Stallinga P, Yassievich IN, Gregorkiewicz T (2008) Nat Photonics 2:105–109
170. Bruhn B, Limpens R, Chung NX, Schall P, Gregorkiewicz T (2016) Sci Rep 6:20538
171. Achstein AW et al (2016) Phys Rev Lett 116:116802
172. Jeong KS, Guyot-Sionnest P (2016) ACS Nano 10:2225–2231
173. Deng Z, Guyot-Sionnest P (2016) ACS Nano 10:2121–2127
174. Lutich AA et al (2010) Nano Lett 10:4646–4650
175. Krahe R et al (2011) Appl Phys Lett 98:063105
186. Wong JI et al (2014) ACS Nano 8:2873–2879
187. Galland C et al (2013) Nano Lett 13:321–328
188. Tisdale WA et al (2010) Science 328:1543–1547
189. Padilha LA et al (2011) ACS Nano 5:5045–5055
190. Galland C et al (2011) Nature 479:203–207
191. Polman A, Atwater HA (2012) Nat Mater 11:174–177
192. Allan G, Delerue C (2009) Phys Rev B 79:195324
193. Cohn AW, Schimpf AM, Gunthardt CE, Gamelin DR (2013) Nano Lett 13:1810–1815
194. Boehme SC et al (2015) Nano Lett 15:3056–3066
195. Nirmal M et al (1996) Nature 383:802–804
196. Efros AL, Rosen M (1997) Phys Rev Lett 78:1110–1113
197. Klimov VI et al (2000) Science 290:314–317
198. Ivanov SA et al (2004) J Phys Chem B 108:10625–10630
199. Klimov VI, Mikhailovsky AA, McBranch DW, Leatherdale CA, Bawendi MG (2000) Science 287:1011–1013
200. Park Y-S, Bae WK, Pietryga JM, Klimov VI (2014) ACS Nano 8:7288–7296
201. Vaxenburg R, Rodina A, Lifshitz E, Efros AL (2016) Nano Lett. doi:10.1021/acs.nanolett.6b0066
202. Istrate E et al (2008) J Phys Chem B 112:2757–2760
203. Kobayashi Y, Pan L, Tamai N (2009) J Phys Chem C 113:11783–11789
204. Robel I, Greshack R, Kortshagen U, Schaller RD, Klimov VI (2009) Phys Rev Lett 102:177404
205. Pijpers IJH, Milder MTW, Delerue C, Bonn M (2010) J Phys Chem C 114:6318–6324
206. Spicinelli P et al (2009) Phys Rev Lett 102:136801
207. Gomez DE, van Embden J, Mulvaney P, Fernèe MJ, Rubinsztein-Dunlop H (2009) ACS Nano 3:2281–2287
208. Park Y-S, Bae WK, Padilha LA, Pietryga JM, Klimov VI (2014) Nano Lett 14:396–402
209. Zhang K, Chang H, Fu A, Alivisatos AP, Yang H (2006) Nano Lett 6:843–847
210. Jha PP, Guyot-Sionnest P (2009) ACS Nano 3:1011–1015
211. Cohn AW, Rinehart JD, Schimpf AM, Weaver AL, Gamelin DR (2014) Nano Lett 14:353–358
212. Schaller RD, Sykora M, Jeong S, Klimov VI (2006) J Phys Chem B 110:25332–25338
213. Nair G, Zhao J, Bawendi MG (2011) Nano Lett 11:1136–1140
214. Mangum BD, Ghosh Y, Hollingsworth JA, Htoo H (2013) Opt Express 21:7419–7426
215. Beyler AP et al (2014) Nano Lett 14:6792–6798
216. Bae WK et al (2013) Nat Commun 4:2661
217. Gao Y, Sandeep CSS, Schins JM, Houtepe AJ, Siebbeles LDA (2013) Nat Commun 4:2329
218. Vaxenburg R, Rodina A, Shabaev A, Lifshitz E, Efros AL (2015) Nano Lett 15:2092–2098
219. Hollingsworth JA (2013) Chem Mater 25:1318–1331
220. Qin W, Liu H, Guyot-Sionnest P (2014) ACS Nano 8:283–291
221. García-Santamaría F et al (2011) Nano Lett 11:687–693
222. Nasilowski M, Spicinelli P, Patriarche G, Dubertret B (2015) Nano Lett 15:3953–3958
223. Bae WK et al (2013) ACS Nano 7:3411–3419
224. Climente JI, Movilla JL, Planelles J (2012) Small 8:754–759
225. Cragg GE, Efros AL (2010) Nano Lett 10:313–317
226. Park Y-S et al (2011) Phys Rev Lett 106:187401
227. Zhao J, Chen O, Strasfeld DB, Bawendi MG (2012) Nano Lett 12:4477–4483
228. Norris DJ, Yao N, Charnock FT, Kennedy TA (2001) Nano Lett 1:3–7
229. Senden T, Rabouw FT, Meijerink A (2015) ACS Nano 9:1801–1808
230. Shabaev A, Rodina AV, Efros AL (2012) Phys Rev B 86:205311
231. Liu H, Guyot-Sionnest P (2010) J Phys Chem C 114:14860–14863
232. Justo Y et al (2013) J Phys Chem C 117:20171–20177
233. Li L et al (2011) J Am Chem Soc 133:1176–1179
234. Droreros N, Seintis K, Fakis M, Gardelis S, Nasiopoulos AG (2015) J Lumin 167:333–338
235. Van Driel AF et al (2005) Phys Rev Lett 95:236804
236. Dohalová K et al (2013) Light Sci Appl 2:e47
237. Protesscru L et al (2015) Nano Lett 15:3692–3696
238. Lodahl P et al (2004) Nature 430:654–657
239. Kress SJP et al (2015) Nano Lett 15:6267–6275
240. Rabouw FT et al (2015) Nano Lett 15:7718–7725
241. Kim S, Fisher B, Eisler H-J, Bawendi M (2003) J Am Chem Soc 125:11466–11467
295. Kuno M, Fromm DP, Hamann HF, Gallagher A, Nesbitt DJ (2001) J Chem Phys 115:1028–1040
296. Smyder JA et al (2014) Phys Chem Chem Phys 16:25723–25728
297. Sher PH et al (2008) Appl Phys Lett 92:101111
298. Jones M, Lo SS, Scholes GD (2008) Proc Natl Acad Sci 106:3011–3016
299. Verberk R, Van Oijen AM, Orrit M (2002) Phys Rev B 66:233202
300. Tang J, Marcus RA (2005) J Chem Phys 123:054704
301. Zhao J, Nair G, Fisher BR, Bawendi MG (2010) Phys Rev Lett 104:157403
302. Frantsuzov PA, Volkán-Kacsó S, Jankó B (2009) Phys Rev Lett 103:207402
303. Volkán-Kacsó S, Frantsuzov PA, Jankó B (2010) Nano Lett 10:2761–2765
304. Voznyy O, Sargent EH (2012) Phys Rev Lett 112:157401
305. Pennycook TJ, McBride JR, Rosenthal SJ, Pennycook SJ, Pantelides ST (2012) Nano Lett 12:3038–3042
306. Brittman S, Garnett EC (2016) J Phys Chem C 120:616–620
307. Boyd GD, Kasper H, McFee JH (1971) IEEE J Quant Electron 7:563–573