Supplement of Atmos. Chem. Phys., 22, 6327–6346, 2022
https://doi.org/10.5194/acp-22-6327-2022-supplement
© Author(s) 2022. CC BY 4.0 License.

Supplement of

An investigation into the chemistry of HONO in the marine boundary layer at Tudor Hill Marine Atmospheric Observatory in Bermuda

Yuting Zhu et al.

Correspondence to: Yuting Zhu (y.zhu@syr.edu) and Xianliang Zhou (xianliang.zhou@health.ny.gov)

The copyright of individual parts of the supplement might differ from the article licence.
S1 Photolysis rate constant calculation

The photolysis rate constants ($f_i$) for NO$_2$, HONO, HNO$_3$ and O$_3$ was calculated combining measured ultra-violet ($UV_{measured}$) light intensity (Eppley TUV radiometer, 295–385 nm) with the tropospheric ultraviolet and visible radiative transfer model (https://www2.acom.ucar.edu/modeling/tropospheric-ultraviolet-and-visible-tuv-radiation-model). The radiative transfer model calculated the gas-phase photolysis rate constants $J_i^{model}$ and light intensity between 295–385 nm ($UV_{model}$) under clear-sky conditions. The location (32.2647° N, 64.8788° W), height (53 m above the sea level), sampling time, total ozone column, and aerosol optical depth at 550 nm were used as model inputs. The total ozone column and aerosol optical depth data were satellite retrieved daily averages that were obtained from the Ozone Monitoring Instrument (Aura satellite) and the Moderate Resolution Imaging Spectroradiometer (Aqua satellite), respectively. Satellite retrieved data were downloaded using NASA’s GIOVANNI tool (https://giovanni.gsfc.nasa.gov/giovanni/). The following equation was used to correct for cloud coverage:

$$J_i = J_i^{model} \times \frac{UV_{measured}}{UV_{model}} \times \frac{UV_{model}^*}{UV_{measured}^*}$$ (S1)

where $UV_{model}^*$ and $UV_{measured}^*$ are the solar noon modeled and measured UV intensities under clear-sky conditions. The difference between $UV_{model}^*$ and $UV_{measured}^*$ is less than 3%.

S2 Aerosol nitrate photolysis rate constant

Aerosol samples collected from Bermuda were used in photochemistry experiments (section 2.2, main text) that determined the production rate constant of HONO from particulate nitrate that is normalized by solar irradiation during tropical noontime ($J_{pNO_3\rightarrow HONO}^N$, s$^{-1}$):

$$J_{pNO_3\rightarrow HONO}^N = \frac{P_{HONO}}{N_{nitrate}} \times \frac{3.0 \times 10^{-7}}{J_{actn}}$$ (S2)

where $P_{HONO}$ (mol s$^{-1}$) is the rate of HONO produced through the photolysis of nitrate on the aerosol filter during the short-period (5 min) light exposure, $N_{nitrate}$ (mol) is the amount of total nitrate in the aerosol sample, and $J_{actn}$ (s$^{-1}$) is the photolysis rate constant of nitrate in an actinometer solution. Equation S2 normalized the particulate nitrate photolysis rate constant to tropical noontime conditions (ground level, solar zenith angle = 0°) where the photolysis rate constant is $\sim 3 \times 10^{-7}$ s$^{-1}$ for aqueous nitrate and $\sim 7 \times 10^{-7}$ s$^{-1}$ for gaseous nitric acid.
Nitrate actinometers were prepared following Jankowski et al. (1999, 2000) in order to monitor the effective UV intensity of the light source employed in the photochemistry experiments. Each chemical actinometer was a 5 ml solution containing 1 mM sodium benzoate and 1 mM sodium nitrate that was added to a quartz cuvette (1 cm pathlength). Upon UV exposure, the photolysis of aqueous nitrate produces OH radicals and the reaction between OH radical and benzoic acid yields salicylic acid. During our experiment, the actinometer was placed under the solar simulator for 5 min, and the produced salicylic acid was quantified by a fluorescence spectrometer (Hitachi L-7480) with an excitation wavelength of 305 nm and an emission wavelength of 410 nm. $J_{\text{actn}}$ was calculated using:

$$J_{\text{actn}} = \frac{[\text{Salicylic acid}]}{[\text{Nitrate}] \times 0.2 \times t}$$

where $[\text{Salicylic acid}]$ is the concentration of salicylic acid produced during the irradiation (mol L$^{-1}$), $[\text{Nitrate}]$ is the concentration of nitrate in the actinometer solution (0.001 mol L$^{-1}$), 0.2 is the production yield of salicylic acid from the reaction of OH radical and sodium benzoate, and $t$ (s) is the irradiation time.
Figure S1: A map of Bermuda Island. The star symbol represents the location of the Tudor Hill Marine Atmospheric Observatory.
Figure S2: Polar wind plots for the (A) spring and (B) summer field campaigns in 2019.
Figure S3: 7-day air mass backward trajectories calculated daily starting at local noontime for the (A) spring and (B) summer field campaigns in 2019. The backward trajectories were calculated with NOAA’S HYSPLIT model [https://www.ready.noaa.gov].
Figure S4: Whisker plots comparing HONO, NOx, pNO$_3^{LPAP}$ concentrations and HONO/NO$_x$ ratios in clean marine air, island-influenced air, and polluted plumes in the North American (NA) air during spring and the North Atlantic Ocean (NAO) air during the summer. The thick lines represent median values, the boxes the 25$^{th}$ and 75$^{th}$ percentiles, and the whiskers the largest value within 1.5 times interquartile range above 75$^{th}$ percentile and the smallest value within 1.5 times interquartile range below 25$^{th}$ percentile.
Jankowski, J. J., Kieber, D. J. and Mopper, K.: Nitrate and nitrite ultraviolet actinometers, Photochem. Photobiol., 70(3), 319–328, doi:10.1111/j.1751-1097.1999.tb08143.x, 1999.

Jankowski, J. J., Kieber, D. J., Mopper, K. and Neale, P. J.: Development and intercalibration of ultraviolet solar actinometers, Photochem. Photobiol., 71(4), 431–440, doi:10.1562/0031-8655(2000)0710431DAIOUS2.0.CO2, 2000.