A quantitative analysis of the direct and indirect costs of nitrogen fixation: a model based on Azotobacter vinelandii

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Nitrogen fixation is advantageous in microbial competition when bioavailable nitrogen is scarce, but has substantial costs for growth rate and growth efficiency. To quantify these costs, we have developed a model of a nitrogen-fixing bacterium that constrains mass, electron and energy flow at the scale of the individual. When tested and calibrated with laboratory data for the soil bacterium Azotobacter vinelandii, the model reveals that the direct energetic cost of nitrogen fixation is small relative to the cost of managing intracellular oxygen. It quantifies the costs and benefits of several potential oxygen protection mechanisms present in nature including enhanced respiration (respiratory protection) as well as the production of extracellular polymers as a barrier to O₂ diffusion, and increasing cell size. The latter mechanisms lead to higher growth efficiencies relative to respiratory protection alone. This simple, yet mechanistic framework provides a quantitative model of nitrogen fixation, which can be applied in ecological simulations.

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Introduction

All organisms need nitrogen for the synthesis of important molecules including nucleic acids and proteins. Nitrogen pervades the environment in the form of N₂ gas, yet it is often a limiting resource on land and in aquatic environments. Nitrogen fixation provides some marine and terrestrial prokaryotes with an ecological advantage but is typically associated with low growth efficiencies and rates (Kuhla and Oelze, 1988a, b; LaRoche and Breitbarth, 2005; Berman-Frank et al., 2007; Goebel et al., 2007), placing significant constraints on the relative fitness of nitrogen fixers and their biogeography. Here we address the question of what underpins the low growth efficiency of nitrogen fixers and develop a quantitative analysis that forms the basis for trait-based ecological and biogeochemical models.

One contribution to the low growth rate/efficiency of nitrogen fixers is the direct cost of reducing dinitrogen. Nitrogen fixation is catalyzed by the nitrogenase enzyme, and uses 8 electrons and at least 16 molecules of ATP to reduce each N₂ molecule to two molecules of NH₃ (Sohm et al., 2011).

\[
\text{N}_2 + 8\text{e}^- + 8\text{H}^+ + 16\text{ATP} + 16\text{H}_2\text{O} \rightarrow 2\text{NH}_3 + \text{H}_2 + 16\text{ADP} + 16\text{Pi} \quad (1)
\]

Although the energetic expense of breaking the dinitrogen triple bond is high, this direct cost may not necessarily be the greatest challenge faced by nitrogen-fixing cells. Nitrogenase consists of an Fe–protein and a Mo–Fe–protein and other metal cofactors whose activity is decreased upon exposure to oxygen (Gallon, 1981). Thus, aerobic nitrogen fixers have developed numerous strategies to protect nitrogenase from oxygen, particularly in oxygenic phototrophs. For example, in Anabaena and Nostoc the nitrogen-fixing cells also become heterocysts, losing the ability to fix carbon and building a thick cell coating, which is less permeable to oxygen (Neilson et al., 1971; Wolk, 1996; Golden and Yoon, 2003) demanding an exchange of metabolites between cells, as do nitrogen fixers in symbiotic relationships with other species. Free-living, unicellular aerobic nitrogen fixers use other strategies, including the separation of oxygenic photosynthesis and nitrogen fixation between day and night (for example, Colón-López et al., 1997; Misra, 1999; Tucker et al., 2001; Berman-Frank et al., 2003; Saito et al., 2011; Großkopf and LaRoche, 2012). Heterotrophic, soil-dwelling nitrogen fixers, such as Azotobacter vinelandii, use elevated rates of
respiration to deplete intracellular oxygen (Poole and Hill, 1997). They also produce an extracellular polymeric substance that impedes oxygen invasion (Sabra et al., 2000), and they maintain larger cell sizes (Post et al., 1982), which likely also reduces the specific oxygen-invasion rate.

Each of these strategies demands that the organism invests resources in nitrogen fixation that could otherwise be used for other activities, such as carbon fixation or biosynthesis, offsetting the ecological advantage of nitrogen fixation. The resulting reduction in growth rate and/or efficiency is the key to understanding the biogeography of nitrogen fixation in the environment (for example, Staal et al., 2003; Monteiro et al., 2010; Dutkiewicz et al., 2012). In current ecological and biogeochemical simulations, this cost is typically empirically imposed (for example, Krishnamurthy et al., 2007; Dutkiewicz et al., 2012; Stukel et al., 2014). Here we present a mechanistic and quantitative approach to modeling the cost of nitrogen fixation and its associated oxygen management. Although the model represents a generalized approach, we focus on the heterotrophic, nitrogen-fixing soil bacterium, *A. vinelandii*, for which extensive, quantitative laboratory characterizations have been published (for example, Post et al., 1982, 1983a, b; Kuhla and Oelze, 1988a, b; Allman et al., 1990; Poole and Hill, 1997; Oelze, 2000; Dixon and Kahn, 2004).

Laboratory cultures of *Azotobacter* (for example, Kuhla and Oelze, 1988a, b) have demonstrated a decrease in growth efficiency at all growth rates as the ambient concentration of oxygen in the medium was increased. This was manifested as an increased carbohydrate cost, which has been interpreted as enhanced energetically uncoupled respiration (exceeding the level required for biosynthesis; Nagai and Aiba, 1972; Kuhla and Oelze, 1988b) to manage intracellular oxygen concentration and maintain functional nitrogenase, hereafter termed ‘respiratory protection’ (Dalton and Postgate, 1969; Poole and Hill, 1997). This is illustrated in Figure 1a (data in circles; lines represent a model discussed later), in which the specific sucrose consumption rate increased with oxygen concentration. The sensitivity to oxygen manifests largely as a ‘maintenance’ effect, increasing overall carbohydrate consumption (Figure 1a) because the cell must be equally clear of oxygen even at very low growth and nitrogen fixation rates. *Azotobacter* encodes and expresses cytochrome bd, which has high oxygen consumption relative to ATP production (Poole and Hill, 1997). The reduction of growth efficiency with increasing oxygen concentration was also reflected in a lower standing stock of protein (Kuhl and Oelze, 1988b) in Figure 1b. At the lowest oxygen concentration (12 μM) there was a sharp change in the slope at a dilution rate of ~ 0.1 per day.

However, it was noted that the increase in respiration is not linear with oxygen concentration, increasing less rapidly at concentrations above 70 μM (Oelze, 2000) suggesting that other protection mechanisms must also be at play. *A. vinelandii* is also observed to increase in cell size in more oxygenated cultures (Post et al., 1982) and to form an external alginate capsule as a barrier to oxygen diffusion, which increases in molecular weight and gluronic acid content with higher oxygen conditions (Sabra et al., 2000).

The thorough characterization of the energetics of *A. vinelandii* in the laboratory (for example, Post et al., 1982; Bühler et al., 1987a, b; Kuhla and Oelze, 1988a, b; Oelze 2000) provides an excellent opportunity to develop a quantitative, mechanistic model of nitrogen-fixer physiology, which we can employ both as a diagnostic tool and in dynamic ecological simulations. Shortly, we will present such a model, but first we consider the general principle that underpins it.

**Materials and methods**

In this section, we describe the core equations of the model. The entire equations are summarized in Supplementary Material S4 and the nomenclature is provided in Supplementary Material S5.
Intracellular oxygen balance

Oxygen management is a key cost for nitrogen fixers that we seek to quantitatively model. First consider the rate of change of the intracellular oxygen, \( Q_{O2} \) (mol O\(_2\) per cell) in a spherical microbe:

\[
\frac{dQ_{O2}}{dt} = P_{O2} + 4\pi k_{O2} (|O2| - |O2|_C) - (R_S + R_m + R_P)
\]  
(2)

Here, \([O2]_C\) and \([O2]_C\) are the environmental and intracellular oxygen concentrations, respectively (mol O\(_2\) m\(^{-3}\)). The first term on the right, \(P_{O2}\) (mol O\(_2\) per cell per s) represents a source from oxygenic photosynthesis. The second term is a source because of transfer across the membranes of cell with the cytoplasmic radius \( r \) (m per cell), governed by the oxygen gradient and the effective diffusivity across the membrane and the effective diffusivity of oxygen in the intra- and extracytoplasmic membrane increased with the intracytoplasmic radius \( r \) (m\(^2\) s\(^{-1}\)). The third term, in parentheses, represents consumption of intracellular oxygen by respiration associated with synthesis \( (R_s)\) including the direct cost of nitrogen fixation, maintenance \( (R_m)\) and respiratory protection \( (R_P)\). \( R_S \) is related to the growth rate of the population, \( \mu \) (s\(^{-1}\)) by

\[
R_S = \mu Q_C Y^{O2\_BIO}_S
\]  
(3)

where \( Q_C \) is the carbon quota (mol C per cell) of the species in question and \( Y^{O2\_BIO}_S \) is the growth yield with respect to oxygen (mol O\(_2\) consumed per mol C biomass synthesized), which can be evaluated from the overall stoichiometry of the reactions (Heijnen and Roels, 1981; Rittmann and McCarty, 2001; see Supplementary Material S1).

As reducing intracellular oxygen concentration is critical for nitrogen fixers, consider the solution for the intracellular oxygen concentration \([O2]_C\) at steady state \( (dQ_{O2}/dt=0): \)

\[
[O2]_C = [O2] + P_{O2} - (\mu Q_C Y^{O2\_BIO}_S + R_m + R_P) \frac{4\pi k_{O2}}{[O2]_C}
\]  
(4)

Oxygenic photosynthesis, \( P_{O2} \), always acts to increase intracellular oxygen concentration along with invasion from the environment, if the external concentration is higher. In contrast, there are numerous strategies to reduce intracellular oxygen levels and protect nitrogenase, as mentioned in the introduction: living in a low-oxygen environment, reducing \([O2]_C\); increasing the efficiency of respiratory oxygen consumption, \( Y^{O2\_BIO, BIO}_S \), and creating thick membranes or mucus layers to reduce the effective diffusivity of oxygen, \( k_{O2} \), into the cell. As carbon quota, \( Q_C \), increases with cell volume \( (r^3) \), increasing cell radius will increase \( R_S \) and reduce \([O2]_C\), as well increasing growth rate \( \mu \) also increase the respiratory oxygen demand. A high maintenance respiration or deliberate respiratory protection, \( R_P \), consumes oxygen. The investment in respiratory protection to reduce the intracellular oxygen concentration to very low levels can be estimated by setting \([O2]_C = 0\) in Equation (4) and re-arranging:

\[
R_P = 4\pi k_{O2} P_{O2} - (\mu Q_C Y^{O2\_BIO}_S + R_m) \frac{4\pi k_{O2}}{[O2]_C}
\]  
(5)

The required \( R_P \) is the difference between sources due to oxygenesis and diffusive invasion, and the demand from growth and maintenance.

A model of A. vinelandii

A. vinelandii employs several of the strategies encapsulated in Equation (4): it employs respiratory protection (see Figure 1). Its radius increases with the environmental oxygen concentration (Post et al., 1982, 1983b; Poole and Hill, 1997). If the respiration rate is proportional to the cell volume, that is, the rate is influenced by \( r^3 \) (which is approximately the case for prokaryotes; Glazier, 2009), and then the cell radius influences the intracellular oxygen concentration in a quadratic manner according to Equation (4). Thus, the influence of the cell radius on oxygen management can be significant. Azotobacter also creates an alginate barrier, as the environmental oxygen concentration increases, and the amount of intracytoplasmic membrane increased with environmental oxygen concentration (Sabra et al., 2000).

Here we extend the simple model of the previous section to develop a more complete representation of a heterotrophic nitrogen fixer (the cell flux model) that we interface with published data on A. vinelandii cultures. This constrains the key parameters, and the model, in turn, provides a quantitative interpretation of the cultures. Extending the simple model of Equation (3), we consider carbon, nitrogen, electron and energy flow in a simplified cell. We combine an idealized biochemical model (Rittmann and McCarty, 2001) that guides the evaluation of growth efficiencies such as \( Y^{O2\_BIO}_S \), with a diffusive representation of the transport of oxygen and other substrates through the cell membrane and surrounding molecular boundary layer (Staal et al., 2003).

Ultimately, the model has two free parameters: \( \epsilon \), the energetic efficiency of cellular metabolism, and \( \epsilon_m \), the efficiency of diffusion of oxygen through the cell membrane. We constrain these parameters using the laboratory data for Azotobacter illustrated in Figure 1, and use the model to infer the cost of nitrogen fixation and the relative effect of different protection mechanisms. Although we focus on Azotobacter, for which systematic data are available, the model is potentially applicable to a broad range of nitrogen-fixing microbes, and we discuss the broader implications in later sections.

The model, depicted schematically in Figure 2, conserves mass, energy (when \( R_P = 0 \)) and electron flow at the individual scale and in the simulated reactor vessel, and assumes that cultures are in steady state. The key rate Equations (6), (7) and (10) (Table 1) describe the cellular-scale conservation of carbohydrate and oxygen, as well as conservation of carbohydrate in the reactor vessel.
For a steady state, conservation of carbohydrate at the cellular scale (Equation (6)) balances uptake of the carbohydrate (CH) substrate, \( V_{CH} \), against synthesis into ‘biomass’ at dilution rate \( D \) (s\(^{-1}\)), respiration to support that synthesis (the second term in the equation), maintenance respiration \( mQ_c \) and additional respiration associated with respiratory protection \( R_P/Y^{O_2,CH}_{N-S} \) (all mol C per cell per s).

An expression for the specific carbohydrate uptake rate, \( C_S \) (mol C in carbohydrate s\(^{-1}\)/mol C in biomass) = \( V_{CH}/Q_c \), can be found by rearrangement of Equation (6):

\[
C_S = \frac{Y^{CH,BIO} D}{P_V} + m + \frac{R_P}{Q_c Y^{O_2,CH}_{N-S}} \tag{11}
\]

Here \( Y^{CH,BIO} \) is the cellular rate of CH consumption required to supply the balanced electron, carbon and energy for biomass synthesis including nitrogen fixation. \( Q_c \) is the cellular quota of carbon (mol C per cell), which is determined as a function of cell volume \( V \) (Bratbak and Dundas, 1984); see Supplementary Material S3 for the calculation of \( V \). \( Y^{O_2,CH}_{N-S} \) is the ratio in which oxygen is consumed relative to carbohydrate for non-synthesis-related respiration (that is, maintenance and respiratory protection). We evaluate \( Y^{CH,BIO} \) and \( Y^{O_2,CH}_{N-S} \) using the method described by Rittmann and McCarty (2001), balancing mass, electron and energy flow at the individual scale (outlined in Supplementary Material S1). The simple model for \( C_S \) in Equation (11) suggests a linear relationship with dilution rate (average growth rate), \( D \), and additional costs due to minimal maintenance and respiratory protection. If the latter is neglected (\( R_P = 0 \)) then Equation (11) represents the classical Pirt model (Pirt, 1982), where \( Y^{CH,BIO} \) is constrained by exact energy balance. In this study, however, we assume that the maintenance rate \( m \) is negligible, as it is typically small relative to respiratory protection in most instances explored here. We note that \( Y^{CH,BIO} \) is a function of the energy transfer efficiency, \( \epsilon \), which characterizes unresolved details of metabolic adaptations (see Results and Discussion, and Supplementary Material S1) and, as such, is a tunable parameter of this model.

The vitality ratio, \( P_V \), represents the fraction of active cells in the culture. Under unfavorable conditions, such as nutrient depletion, cells can survive in dormancy with low metabolic rate.

Table 1 Fundamental relationships of the cell flux model

| Relationship | Expression |
|-------------|------------|
| Cellular conservation of carbon (mol C cell\(^{-1}\) s\(^{-1}\)) | \( \frac{dQ_{CH}}{dt} = V_{CH} - \frac{DQ_c Y^{CH,BIO}}{P_V} - mQ_c - \frac{R_P}{Y^{O_2,CH}_{N-S}} \) |
| Cellular conservation of oxygen (mol O\(_2\) cell\(^{-1}\) s\(^{-1}\)) | \( \frac{dQ_{O_2}}{dt} = V_{O_2} - \frac{DQ_c Y^{O_2,BIO}}{P_V} - mQ_c Y^{O_2,CH}_{N-S} - R_P \) |
| Cellular oxygen uptake (mol O\(_2\) cell\(^{-1}\) s\(^{-1}\)) | \( V_{O_2} = 4\pi r \kappa_{O_2} (|O_2| - |O_2|_c) \) |
| Effective diffusivity of oxygen (m\(^2\) s\(^{-1}\)) | \( \kappa_{O_2} = \frac{K_0 \left( L_s + r \right)}{L_s / \varepsilon_m + r} \) |
| Conservation of carbon in the reactor (mol C m\(^{-3}\) s\(^{-1}\)) | \( \frac{d[CH]}{dt} = D([CH]_{IN} - [CH]) - V_{CH}XP_V \) |

Supplementary Material S5 provides the definition of each parameter.
(Lennon and Jones, 2011). In continuous culture, there are also dormant or non-vital cells (Postgate, 1973), possibly because of nutrient limitation, and the division rate reflects the average from growing and non-growing cells. The vitality ratio has a significant impact on the growth efficiency as the respiration rate of the non-vital cells is observed to be low (Postgate, 1973). Here we parameterize the fraction of non-vital cells as a function of growth rate based on empirical data of Postgate (1973) for cultures of *Klebsiella aerogenes*:

\[
P_v = 0.18547 \ln(D) + 2.75176 \quad (12)
\]

At present, there are no explicit data on the viability rate of *Azotobacter*, but incorporating this effect as calibrated for *Klebsiella* significantly improves the simulation. Specifically, this mechanism reduces the effective yield of the simulation with the dilution rate (that is, increases the slopes of the \(C_S\) versus \(D\) curves).

Respiratory protection appears as an additional ‘maintenance’ term in Equation (11), consistent with that observed as the oxygen stress increases in *A. vinelandii* (increasing the total carbohydrate consumption with oxygen in data on Figure 1a). Higher environmental oxygen concentrations increase the invasion rate, requiring more vigorous respiratory protection and reducing the overall growth efficiency. In order to fix nitrogen, the cell must keep the intracellular oxygen concentration very low (that is, \([O_2]_{c}=0\) at all rates of growth (dilution). Thus, we can constrain \(R_p\) by estimating the rate of oxygen invasion into the cell using Equation (13), derived from Equations (7) and (8) from Table 1 under the steady-state assumption.

\[
\frac{R_p}{Q_C} = \frac{4\pi \kappa_{O_2}[O_2]}{Q_C} - \frac{D Y_{O_2:BIO}^{S}}{P_v} - \frac{m Y_{O_2:CH}^{N-S}}{PR} \quad (13)
\]

Here \(Y_{O_2:BIO}^{S}\) represents moles of oxygen gas consumed per mole of carbon biomass production for synthesis. Although energy balance can provide some constraint, the realized efficiency of energy transfer depends upon unresolved details of the specific metabolic pathways utilized by the organism and is encapsulated by the energy efficiency parameter, \(\varepsilon\) (see Supplementary Material S1).

The diffusive transfer of oxygen from the ambient environment into the cell is described by Equations (8) and (9), following Staal et al. (2003), where the effective diffusivity, \(\kappa_{O_2}\), depends on the diffusivity in water, the cytoplasmic radius, \(r\), and the thickness and relative diffusivity of the cell membrane layers (\(L_m\) and \(\varepsilon_m\)). We impose a positive correlation between the cell radius of *Azotobacter* and the ambient oxygen concentration based on the empirical data of Post et al. (1982). Here we assume a fixed ratio between the cell membrane thickness and cell radius in order to express the effect of increasing intracytoplasmic-membrane vesicles with the cell radius (Post et al., 1982). As the diffusivity of oxygen in water is well known, we are left with a second control parameter, \(\varepsilon_m\). When the invasion of oxygen exceeds the minimum energy demand, the model invokes respiratory protection sufficient to consume all the excess oxygen flux.

Substituting for \(R_p\) in Equation (11) with Equation (13) provides a quantitative, mechanistic model for the growth efficiency of nitrogen-fixing *Azotobacter* that incorporates both direct and indirect costs. Using conservation of carbohydrate in the reactor vessel, Equation (10) in Table 1, and assuming that at the steady state the carbohydrate is drawn down to a low, subsistence concentration (that is, \([\text{CH}] \ll [\text{CH}]_S\)), we find an expression for the standing concentration of protein in the reactor:

\[
[PR] = \frac{D [\text{CH}]_S}{C_S P_v} Y_{PR:BIO} \quad (14)
\]

where \(Y_{PR:BIO}\) is the ratio of carbon in protein to carbon in total biomass. We have used 1:1.32 for this ratio based on the laboratory study of Bühler et al. (1987a).

Considerations of mass, electron and energy balance at the individual and reactor scales, along with an estimate of the rate of invasion of oxygen through the cell wall, allowed us to fully describe the system and predict the specific carbohydrate consumption rate, \(C_S\) and concentration of protein in the reactor [PR].

### Results and discussion

#### Direct and indirect costs of nitrogen fixation

The model described above was brought into consistency with the data of Kuhla and Oelze (1988b; data points in Figure 1) by tuning the two control parameters (that is, \(\varepsilon_m = 7.9 \times 10^{-4}\) and \(\varepsilon = 0.22\)) and approximating the energetics of one sucrose molecule as that of two glucose molecules. The simulations qualitatively and quantitatively capture the key features of the laboratory cultures. It identifies the transition between energy-limited and oxygen-controlled regimes in the experiment, confirms that *Azotobacter* is using several strategies to manage oxygen, quantifies their relative contributions and quantifies the allocation of carbon resources, showing that indirect cost of oxygen management far exceeds the direct cost of nitrogen fixation.

#### Transition between energy limitation and oxygen management regimes

The solid curves in Figure 1 depict the optimized model solutions for the carbohydrate consumption rate and protein concentration against dilution rate for different oxygen concentrations along with laboratory data in Kuhla and Oelze (1988b). In most cases, the diffusive oxygen supply exceeds the demand for synthesis and the respiratory protection is present, increasing specific sucrose consumption at all dilution rates. However, at higher dilution rates and low ambient oxygen
concentrations ([O₂] = 12 μM), respiratory protection is no longer necessary. There growth is oxygen-limited (see equation (3)), and the minimum respiration limit is appropriate. This leads to a sharp transition in the appropriate curves on Figure 1 particularly so for the protein standing stock (Figure 1b); it represents the transition between energy-limited and oxygen management regimes. The model captures and explains this transition.

**Multiple oxygen protection strategies.** The simulations capture the trend of increasing the carbohydrate consumption rate both with dilution rate and oxygen concentration (Figure 1a, solid lines). When oxygen management is not a major control, for example, the minimum respiration case at [O₂] = 12 μM, the increase in the carbohydrate consumption rate with dilution rate is mainly caused by increasing biomass production, nitrogen fixation and balanced respiration. At higher oxygen concentrations, there is an overall increase in carbohydrate consumption because of respiratory protection (that is, appears as a ‘maintenance’ effect). High dilution rates also increase carbohydrate consumption because of the increased proportion of viable cells, which need respiratory protection for fixing nitrogen, whereas non-viable cells do not.

Figure 3 shows data and model solutions for Cₛ as a function of oxygen concentration at a fixed dilution rate. The dotted black line indicates the required specific sucrose consumption rate if the cells respond only by respiratory protection to increasing ambient oxygen concentrations. As Oelze (2000) pointed out, this linear response is inconsistent with the data (Figure 3, circles) and demands additional mechanisms. This nonlinearity is captured by the model if the specific invasion rate of oxygen is reduced by an increase in cell radius (r+Lₒ) or a reduction in the effective diffusivity of oxygen invasion (κₒ₂) under higher oxygen concentrations, either by thickening the cell membrane or developing an alginate capsule at higher oxygen concentrations (as observed by Post et al., 1982; Sabra et al., 2000). A relatively minor change in effective cell diameter is sufficient. In the solid grey line of Figure 3 an increase in cell diameter of a factor of 1.4 (1.76–2.41 μm) as oxygen concentration increased from 12 to 192 μM brought the model into consistency with the observed data. This is consistent with the increase in size by a factor of 1.6 observed by Post et al. (1982) over a similar range of oxygen concentration in the lab.

*Azotobacter* can also produce an alginate capsule, which decreases the diffusivity on the cell boundary layer (Sabra et al., 2000), although this phenomenon is not always reported in the laboratory studies. We represent the development of an alginate capsule by altering the effective diffusivity, and imposing a carbohydrate cost for alginate production (dashed black line in Figure 3; for details see Supplementary Material S2). Although sucrose expenditure is increased in order to achieve larger cell radius or alginate layer, the model clearly shows these strategies to significantly improve growth efficiency at higher oxygen concentrations, relative to pure respiratory protection (that is, the solid grey and dashed black are significantly lower than the dotted black line in Figure 3).

The model confirms that multiple oxygen protection strategies must be at play in *A. vinelandii*. At higher oxygen concentrations, changes in cell volume and/or alginate production supplement respiratory protection and significantly increase growth efficiency, whereas at lower concentrations, respiratory protection alone is as effective.

**Allocation of resources.** The qualitative and quantitative fit to the laboratory data suggest that the model is a suitable tool for quantitative interpretation of resource allocation by *A. vinelandii* in different oxygen environments. The analysis, shown in Figure 4, reveals that in low ambient oxygen concentrations the major respiratory cost is synthesis (Figure 4a). In an oxygenated environment, the greatest carbohydrate cost is for respiratory protection (Figure 4b), and the direct costs of nitrogen fixation are quite small (yellow and red portions, Figure 4b).

We note that the whole-cell energy-transfer efficiency for *Azotobacter*, ε = 0.22, inferred from the model fit is rather low relative to comparable simulations of other organisms growing on ammonium and glucose, where ε = 0.54, using data from Heijnen and Roels (1981). Even when oxygen is low and respiratory protection is not in play (Figure 4a), the efficiency of carbohydrate utilization for synthesis remains relatively low. *A. vinelandii* expresses cytochrome bd, which has high oxygen consumption relative to ATP production (Poole and

![Figure 3](image-url)
hill, 1997), trading off low energy efficiency against the ability to manage oxygen for nitrogen fixation. There may also be a high cost in protein turnover, as nitrogenase is so sensitive to even low oxygen concentrations.

**Broader Context**

**Nitrogen fixation and oxygen management**

The comprehensive physiological and energetic data sets characterizing *A. vinelandii* have not been obtained for most nitrogen fixers. However, the model indicates that other types of nitrogen fixers may employ similar oxygen management strategies despite their differences in their sizes and physiology (Table 2). The model has two free parameters that were constrained by minimizing model-data differences: $\epsilon$ reflects the efficiency of energy production and $\epsilon_m$ encapsulates the permeability of the cell membrane layers to oxygen. Assuming that this low energy-transfer efficiency is common among nitrogen fixers, we can apply the cell flux model to consider the oxygen management strategies of other nitrogen fixers. Table 2 illustrates the variety of strategies and their occurrence among several nitrogen-fixing species: terrestrial and aquatic, heterotrophic and phototrophic. We note that several of these species are phototrophic; therefore, the energetic constraints may be rather different during the day. However, in the dark (when some are fixing nitrogen) the cell flux model as developed for *Azotobacter* may be informative.

*Crocosphaera* is a free living, unicellular marine nitrogen fixer with a typical cell radius of 1 μm (estimated from Dron et al., 2012). Großkopf and LaRoche (2012) found that ~60% of respiration by *Crocosphaera* is non-essential (that is, respiratory protection) at 186 μM O$_2$. We applied the cell flux model framework to a simulation of *Crocosphaera* in the dark (they fix nitrogen at night) with a growth rate of 0.23 per day (Großkopf and LaRoche, 2012).

![Figure 4](attachment:image.png) **Figure 4** Allocation of carbohydrate resource (as measured by the contribution to specific carbohydrate consumption rate) at (a) 12 μM O$_2$ and (b) 192 μM O$_2$. Solid circles indicate data from Kuhla and Oelze, 1988b.

| Strategy            | Size change | Extracellular polymeric substance                          | Respiratory protection |
|---------------------|-------------|------------------------------------------------------------|------------------------|
| *Azotobacter*       | Yes$^c$     | Alginate$^b$                                               | Yes$^c$                |
| *Anabaeo* and *Nostoc* | Yes: heterocysts$^d$ | Highly likely: polysaccharide layer$^e$ | —                      |
| *Crocosphaera*      | —           | Possible: EPS$^f$                                           | —                      |
| *Trichodesmium*     | Possible: colony$^b$ | Possible: envelope-related genes$^g$ | Possible$^i$           |

References:
- Post et al. (1982).
- Sabra et al. (2000).
- Pool and Hill (1997).
- Kumar et al. (2010).
- Maldener and Muro-Pastor (2010).
- Dron et al. (2012).
- Großkopf and LaRoche (2012).
- Capone et al. (1997).
- Yang et al. (2007).
- Berman-Frank et al. (2001).
and 100% vitality. To balance the books, the model suggests that *Crocosphaera* must have a very low effective diffusivity for oxygen (that is, a low $\varepsilon_m$), $\sim 1/57$ that of the cell membrane layer of *A. vinelandii* (in the absence of alginate). Such a low $O_2$ permeability could result from either a thickened cell wall or an extracellular polymer layer analogous to alginate in *Azotobacter*. The latter is consistent with the significant amounts of extracellular polymeric substances observed in association with *Crocosphaera* (Dron et al., 2012).

Other organisms use specialized heterocyst cells with very thick cell membranes to exclude oxygen. The cell wall of heterocysts of *Anabaena* has an effective diffusivity of $\sim 1.18 \times 10^{-12}$ m$^2$ s$^{-1}$ for oxygen (estimated from Walsby (1985)). Simulations with the cell flux model assuming this diffusivity, along with a cell radius of 2.89 μm (estimated from Walsby (1985)) and 100% vitality, we find that at typical saturated oxygen concentrations of 225 μM $O_2$ (Post et al., 1982), this organism should not require respiratory protection for growth rates above $\sim 0.10$ per day. Its maximum growth rate is $\sim 0.75$ day (Fogg, 1944).

One strategy of *A. vinelandii* is to increase cell size, therefore reducing surface to volume ratio and the specific invasion rate of oxygen. Forming rafts, with strands of cells bundled together, as seen in the aquatic species *Trichodesmium* and *Aphanizomenon*, may have the same effect. Under saturated oxygen concentrations 225 μM $O_2$ (Post et al., 1982), cell radius of 2.17 μm (estimated from Bidle and Falkowski (2004)), growth rate of 0.23 per day (Prufert-Bebout et al., 1993) and 100% vitality, our model suggests that for an isolated trichome, $\sim 97\%$ of total respiration would be required for respiratory protection. However, if the colony radius triples because of bundling, the central trichomes are shielded from oxygen with an analogy to increasing the radius of individual cells; the surface to volume ratio is reduced such as to increase efficiency with $\sim 75\%$ of respiration expended on oxygen management. With colony radius six times that of a single cell, no respiratory protection is needed for the central trichomes. Hence, the formation of bundles may provide a significant enhancement of growth efficiency for *Trichodesmium* and other marine nitrogen fixers.

Finally, we note that recent studies find heterotrophic nitrogen fixers in various ocean environments (reviewed in Zehr et al., 2003; Riemann et al., 2010), although the exact manner of their lifestyle is yet to be determined. Such organisms might be considered marine analogs of *Azotobacter*. It might seem surprising that such organisms could thrive in the open ocean: given the energetic cost of oxygen management, it is logical that most marine nitrogen fixers would be photoautotrophic or photoheterotrophic. The high cost of oxygen management in the highly aerobic surface ocean, along with the potential paucity of sufficiently rich organic carbon substrates in the dissolved form, makes it hard to envisage heterotrophic nitrogen fixation as a viable strategy for a free-living cell. However, heterotrophic nitrogen fixation can be favored in certain oxygen conditions. With growth rate of 1 per day, our *Azotobacter* study suggests a sweet spot of $\sim 3\%$ $O_2$ saturation at which the organisms are neither oxygen-limited nor poisoned. Such conditions may occur in regions of low oxygen near oxygen minimum zones, or on the microscale in organic particles creating viable habitats. In addition, even under much higher oxygen pressure, given sufficient quantities of carbohydrate substrates, aquatic heterotrophic organisms may carry out nitrogen fixation, as observed recently in *Pseudomonas stutzeri* (Bentzon-Tilia et al., 2015). Alternatively, symbiosis may lead to fruitful sources of high-energy substrates. Although the precise nature and biogeoography of heterotrophic marine nitrogen fixers remain elusive at present, we suggest that the cell flux model developed here may prove to be a useful tool for understanding and modeling these organisms.

**Conclusions**

To examine the growth efficiency of nitrogen-fixing bacteria, we have developed a ‘cell flux model’, constrained by mass, redox and energy balance, to simulate *A. vinelandii* in nitrogen-fixing conditions. The model qualitatively and quantitatively captures the trends in specific carbohydrate consumption, respiration and the standing stock of protein as a function of growth rate and ambient oxygen concentration. It indicates that in oxygenated conditions the direct cost of nitrogen fixation has a limited impact on the growth efficiency of *Azotobacter*, relative to the major investment in respiratory protection. *Azotobacter* also employs other strategies to protect nitrogenase from oxygen including an increase in cell radius and the production of an alginate capsule, both of which increase efficiency relative to respiratory protection alone. The strategies for oxygen management employed by *Azotobacter* are common with many nitrogen fixers, aquatic and terrestrial, heterotrophic and autotrophic. The model has sufficient detail to quantitatively capture many aspects of the energetics and lifestyle of *Azotobacter* and other nitrogen fixers, yet it remains simple, transparent and computationally cheap. We suggest that it can form the basis for a general physiological model with which to simulate and interpret the relative fitness of microbes of a variety of functionality in biogeochemical and ecological simulations.

**Conflict of Interest**

The authors declare no conflict of interest.
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