Review Article

Molecular Regulation of the Mitochondrial F\textsubscript{1}F\textsubscript{o}-ATPsynthase: Physiological and Pathological Significance of the Inhibitory Factor 1 (IF\textsubscript{1})

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In mammals, the mitochondrial F\textsubscript{1}F\textsubscript{o}-ATPsynthase sets out the energy homeostasis by producing the bulk of cellular ATP. As for every enzyme, the laws of thermodynamics command it; however, it is privileged to have a dedicated molecular regulator that controls its rotation. This is the so-called ATPase Inhibitory Factor 1 (IF\textsubscript{1}) that blocks its reversal to avoid the consumption of cellular ATP when the enzyme acts as an ATP hydrolase. Recent evidence has also demonstrated that IF\textsubscript{1} may control the alignment of the enzyme along the mitochondrial inner membrane, thus increasing the interest for the molecule. We conceived this review to outline the fundamental knowledge of the F\textsubscript{1}F\textsubscript{o}-ATPsynthase and link it to the molecular mechanisms by which IF\textsubscript{1} regulates its way of function, with the ultimate goal to highlight this as an important and possibly unique means to control this indispensable enzyme in both physiological and pathological settings.

1. Introduction

The F\textsubscript{1}F\textsubscript{o}-ATPsynthase is an H\textsuperscript{+}-pumping ATPase evolutionary specialized in synthesizing ATP by using an H\textsuperscript{+} gradient generated across a biological membrane. It is present in bacteria and intracellular organelles such as chloroplasts and mitochondria. In these, the enzyme is hosted within the inner membrane as part of the OXPHOS, where it couples the transport of H\textsuperscript{+} from the intermembrane space into the matrix with the synthesis of ATP, guaranteeing the supply of energy to biological processes, since the majority of cellular ATP is generated by the mitochondrial F\textsubscript{1}F\textsubscript{o}-ATPsynthase.

The molecular structure, catalytic mechanism, and regulation of the mitochondrial F\textsubscript{1}F\textsubscript{o}-ATPsynthase were described by the seminal work of the Nobel Laureates Mitchell, Boyer and Walker, that revealed its complexity and the functional steps that drive the synthesis of ATP.

Besides its role as energy producer, the mitochondrial F\textsubscript{1}F\textsubscript{o}-ATPsynthase is also essential for the maintenance of the mitochondrial membrane potential ($\Delta\Psi_m$) [1]—which is crucial for the import of proteins into the organelle [2]—and cristae structure [3]. In addition, it is fundamental for an optimal supramolecular organization of the respiratory chain [4], and for regulating the mobilization of cytochrome c during apoptosis [5].

In animals and plants, the mitochondrial F\textsubscript{1}F\textsubscript{o}-ATPsynthase is molecularly regulated by an endogenous, nuclear-encoded polypeptide, the ATPase Inhibitory Factor 1 (IF\textsubscript{1}). IF\textsubscript{1} is primarily responsible for inhibiting the hydrolysis of ATP by the ATP synthase [6], an event that occurs when the electrochemical proton gradient across the mitochondrial inner membrane is lost (e.g., during hypoxic/ischaemic conditions), and the enzyme reverses to restore $\Delta\Psi_m$ [7]. A number of studies have shown that IF\textsubscript{1} is also involved in the regulation of the oligomeric state of the F\textsubscript{1}F\textsubscript{o}-ATPsynthase, by facilitating the enzyme’s dimerization via a molecular link between two F\textsubscript{1} domains [8]; for this reason, it is also implicated in the remodelling of cristae structure [9], and consequently in the regulation of mito-ultrastructure and morphology.
The interest for this inhibitor, or regulator—as we like to consider it—stems from many reasons; among them, the evidence for a key role in pathologies is the most meaningful but the less explored. IF₁ overexpression is reported in human carcinomas [10], differences in the ratio of expression between IF₁ and the F₁Fₒ-ATP synthase are related to changes in cellular responses to ischaemia/reperfusion injury [11, 12], and its absence is recorded in a rare form of mitochondrial myopathy called Luft’s disease [13, 14].

Despite this, IF₁ seems underconsidered in pathologies whose etiology correlates with defective mitochondrial F₁Fₒ-ATP synthase. Here, we will explain why the interaction between IF₁ and F₁Fₒ-ATP synthase is important, and why the quality of cellular bioenergetics depends on it.

2. Molecular Structure and Catalysis of the F₁Fₒ-ATP synthase

The mitochondrial F₁Fₒ-ATP synthase is the smallest rotary motor in nature. It is a multisubunit complex (~5,000 aminoacid residues, with a mass of ~600 kDa) consisting of an intrinsic membrane domain, Fₒ (~1,500 aminoacids), and a globular catalytic domain, F₁ (~3,500 aminoacids), which extends into the mitochondrial matrix (Figure 1). In mammals, the enzyme contains 15 different subunits, nine of which form the Fₒ domain (a, A6L, b, c, d, f, F₆, e, and g), while the F₁ domain is instead composed of only six (α, β, γ, δ, ε, and OSCP) [15–17]. The Inhibitory Factor 1 (IF₁) is often regarded as the 16th subunit, although we will learn in this review that the protein is far more correctly defined as its endogenous regulator.

The enzyme could be ideally divided into 4 principal subdomains: the catalytic headpiece (α₃β₃), hosting the three catalytic sites for ATP synthesis (one in each β subunit), a proton channel (αδεγ), and two stalks, the central rotor (γδε), and the peripheral stator (bd(F₆)OSCP) that link the first two subdomains together. While protons flow through the Fₒ channel from the intermembrane space into the matrix, a rotation of the stator inside the catalytic headpiece is induced, allowing a cyclic change in β-subunits conformation and the synthesis of ATP (N.B. Subunits A6L, e, f, and g are omitted in the scheme).

Figure 1: Oxidative phosphorylation and the mammalian F₁Fₒ-ATP synthase. (a) Scheme of the mitochondrial OXPHOS: it is composed of five complexes, which couple the generation of a proton motive force through the mitochondrial inner membrane (IMM) with ATP synthesis. The first four complexes form the electron-transport chain (ETC), which catalyses the oxidation of NADH and FADH₂ to NAD⁺ and FAD respectively, with the associated reduction of molecular oxygen, to which electrons are transferred, to water. During the process, protons are translocated against a gradient in the intermembrane space by complexes I, III, and IV; the generation of a proton electrochemical potential (ΔμH⁺), also called proton motive force (pmf), is achieved, driving the ATP synthesis, which is catalyzed as the final step by the F₁Fₒ-ATP synthase (Complex V). (b) Diagram of the structure of mammalian F₁Fₒ-ATP synthase. We can divide the enzymatic complex into 4 principal subdomains: a catalytic headpiece (α₃β₃), the Fₒ domain (αδεγ), and two stalks, the central rotor (γδε) and the peripheral stator (bd(F₆)OSCP) that link the first two subdomains together. While protons flow through the Fₒ channel from the intermembrane space into the matrix, a rotation of the stator inside the catalytic headpiece is induced, allowing a cyclic change in β-subunits conformation and the synthesis of ATP (N.B. Subunits A6L, e, f, and g are omitted in the scheme).
The crystal structure of the mitochondrial F$_1$-ATP synthase, extracted from bovine heart mitochondria, was revealed at the beginning of the Nineties [19, 20]. In this structure, F$_1$ appears like a flattened sphere of 80 Å in height and 100 Å in width, with three α- and three β-subunits arranged alternately forming a cylinder around the coiled-coil structure of the γ-subunit. Sequences of α- and β-subunits are 20% identical [20], and this homology is reflected in their similar folds. All α-subunits have almost the same conformation, while the three β-subunits adopt three different tertiary structures and are in three diverse nucleotide-bound states: the first, named β$_{TP}$, hosts the binding site for ATP; the second, called β$_{DP}$, binds with high-affinity ADP and P$_i$; the third, β$_E$, does not efficiently bind any nucleotide. This asymmetry in the conformation and nucleotide occupancy of β-subunits supports the binding change mechanism of catalysis theorized by Boyer in the early 1970s, and fully developed in 1993 [21, 22].

According to this model, the three catalytic sites can be in three different conformations at any given time. β$_{TP}$ and β$_{DP}$ have, respectively, a “tight” (T) and a “loose” (L) form, in which ADP and P$_i$ are converted into ATP, while the “open” conformation (O), adopted by β$_E$, permits the release of the newly formed ATP and the acceptance of another molecule of ADP. ADP + P$_i$ are theorized to enter into an O β-subunit, which then assumes an L conformation allowing the synthesis of ATP. After the synthesis, subunit β is in the T conformation, before assuming the O conformation in order to release ATP. The sequential interconversion between these different conformations, driven by the rotation of the central γ-subunit relative to the (αβ)$_3$ barrel, enables the catalysis.

Unfortunately, there is less structural information on the F$_o$ domain of ATP synthase than the F$_1$ complex. In the ring, subunits c, whose structure was determined by both NMR spectroscopy [23, 24] and X-ray crystallography [25], folds as a hairpin composed by two transmembrane α-helices linked by a polar loop; part of the ring is in contact with subunit a, formed presumably by five transmembrane α-helices and containing two proton half-channels that do not span the membrane [26].

To briefly describe the functioning of the F$_1$F$_o$-ATP synthase and its rotational catalysis, we must divide it into two parts: a central “rotor” (F$_1$γε-F$_o$ε$_{10}$) and a surrounding “stator” (F$_1$α$_3$β$_3$δ$_3$ε$_{10}$). The driving force for the rotation is generated by the electron-transport chain (ETC) and is based on the magnitude of the proton electrochemical potential across the inner membrane; this gradient allows H$^+$ to flow through the F$_o$ domain (the c-ring) causing the rotation of the whole “rotor.” The precession of the γ-subunit, which contacts only one β-subunit at a time, induces cyclic changes in the “stator” so that ATP can be synthesized.

The rotational mechanism is extremely complex and depends on the structures of subunits a and c; the latter contain, in the middle of their C-terminal α-helix, a critical aminoacid residue, D61, which can either be in a protonated (lipophilic) or unprotonated (hydrophilic) form. Protonation and deprotonation of this aspartic acid residue is at the basis of the c-ring rotation dictating the membrane affinity of the single subunits. This occurs only in an aqueous environment and is realized when subunit c is in contact with subunit a, and the aspartate residue is hosted in one of the proton half-channels (see [27] for further details).

In respiring mitochondria, the H$^+$ motive force ensures the entrance of H$^+$ residing in the intermembrane space into the cytosolic half-channel. The [H$^+$] is more than 25 times higher on the cytosolic side than on the matrix side, and the ΔΨ$_m$ of 140–180 mV increases the [H$^+$] near the mouth of the cytosolic half-channel, thus resulting in protonation of the enclosed aspartate. Once the key residue is protonated, the c-ring can rotate clockwise (looking from the intermembrane space side), and the D61 of a new subunit c enters the matrix half-channel of subunit a, releasing a H$^+$ into the mitochondrial matrix. Thus, the difference in [H$^+$] and potential on the two sides of the membrane leads to different probabilities of protonation through the two half-channels, which yields directional rotational motion. The coupling between the rotation of the c-ring and the conformational changes in the (αβ)$_3$ barrel is guaranteed by the tight link of the ring with the central stalk; while the c-ring rotates, the γ-subunit turns inside the αβ$_3$-hexamer, whose rotation is blocked by the presence of the peripheral stalk. Thus, the proton-gradient-driven rotation of the c-ring drives the rotation of the γ-subunit, which in turn promotes the synthesis of ATP through the binding-change mechanism. A highly conserved acidic cluster sequence in the C-terminal helical domain of the β-subunit (the DELSEED motif) is thought to be essential for ATP synthesis by coupling catalysis and rotation [28]. After the binding of an adenosine nucleotide in the catalytic site, the C-terminus is lifted up to the nearly immobile N-terminal part of the protein, and the β-DELSEED sequence is moved in contact with the γ-subunit, probably allowing the coupling of the γ-subunit torque with the rotation of the F$_1$ domain (this hypothesis is still controversial [29]).

Each 360° rotation of the γ-subunit leads to the synthesis and release of three molecules of ATP. The number of subunits in the c-ring, which ranges from 8 (in bovine and probably in all vertebrates and invertebrates F$_o$ domain [30]) to 15 (in Spirulina platensis [31]), determines the number of protons that are required to generate a molecule of ATP.

### 3. Localization and Regulation of the Enzyme

Mitochondria are the sites where cellular energy is most abundantly produced, due to the constant activity of the mitochondrial F$_1$F$_o$-ATP synthase. Although publications have suggested that it may also be localized on the plasma membrane [32, 33], we shall be exclusively discussing and referring to that embedded in the mitochondrial inner membrane.

As the F$_1$F$_o$-ATP synthase is a reversible nanomotor, it can also hydrolyze ATP by translocating H$^+$ from the matrix into the intermembrane space (an event that vigorously occurs when the enzyme is in isolation [34]). It does so to maintain the mitochondrial membrane potential (ΔΨ$_m$) at a suboptimal level during deenergized conditions that occur...
when respiration is impaired by defects in the activity of the ETC, or when the mitochondrial inner membrane is leaky due to alterations in its structural integrity [35]. The \( \Delta \Psi_m \) is not only important for ATP production, but also for mitochondrial protein import and assembly [2]. Disruption of the \( \Delta \Psi_m \) is therefore implicated in various apoptotic phenomena (see [36]), being its maintenance crucial for cell viability.

The reversal of the F\(_1\)F\(_0\)-ATP synthase is avoidable in eukaryotes and the enzyme must be controlled to prevent futile hydrolysis of ATP when the transmembrane proton electrochemical gradient collapses. Only facultative anaerobic bacteria employ this method for generating a vital proton electrochemical gradient in the absence of oxygen [37].

When reversal of the F\(_1\)F\(_0\)-ATP synthase does occur, the depletion of cellular ATP can be more or less severe depending on the energy requests of the tissue, but in organs with high ATP demand, like brain or skeletal muscle, or in case of augmented ATP request, cellular demise is tangibly accelerated.

Apart from pathological states, repression or upregulation of ATP synthesis normally occurs in physiological conditions when intracellular ATP levels are, respectively, sufficiently high or too low. It has been calculated that, in eukaryotic cells, the rate of ATP utilization changes by a factor of 5–10 [38], such as during exercise and/or acclimatization.

Several are the mechanisms by which the activity of the F\(_1\)F\(_0\)-ATP synthase is regulated: (a) transcriptional factors [39, 40]; (b) translational control [41–43]; (c) modulation of the electron transport chain or the citric acid cycle [44, 45]; (d) ADP inhibition [37] and (e) regulatory proteins, such as IF\(_1\). Although recent evidence has also suggested that the oncoprotein Bcl-XL interacts with the F\(_1\)F\(_0\)-ATP synthase [46, 47], IF\(_1\) is the only molecular regulator of the enzyme characterized both biochemically and functionally. Nonetheless, other proteins which directly interact with the enzyme have been identified, such as factor B [48], essential for ATP synthesis and implicated in the regulation of the F\(_1\)F\(_0\)-ATP synthase oligomerization, CaBII [49], which may up-regulate the enzyme in response to increased cytoplasmic Ca\(^{2+}\), and finally S100A1 [50], which has been found to enhance the enzymatic performance in cardiac muscle.

### 4. The Inhibitory Factor 1 (IF\(_1\))

IF\(_1\) was discovered in 1963 by Pullman and Monroy [6] in mitochondria from bovine hearts (a schematic representation of the structure of bovine IF\(_1\) is reported in Figure 2(a)). To date, IF\(_1\) homologues have been isolated from other mammals (e.g., rat [51] and human [52]), yeast (Saccharomyces cerevisiae [53] and Candida utilis [54]) and plants [55]. IF\(_1\) is a small, basic, heat-stable protein of approximately 10 kDa (in mammals the mature form of the polypeptide is composed of a number of amino acid residues which ranges from 81 in human, chimpanzee, dog and mouse, to 84 in cow). It is predominantly compartmentalized inside the mitochondrial matrix (Figure 2(b)), although studies have proposed that IF\(_1\) is also present in the cytosol and on the plasma membrane [56], as well as secreted into the extracellular environment, where it is implicated in the modulation of the activity of endothelial cells [32]. Intriguingly, in this very extramitochondrial localization, a role for hepatic HDL-cholesterol and triglyceride metabolism was also proposed [32, 57].

Figure 2: IF\(_1\): structure and intracellular localization. (a) Schematic representation of bovine IF\(_1\). The mature protein is composed of 84 residues and is \( \alpha \)-helical along most of its length; an amine-terminal prescence of 25 amino acids represents the mitochondrial targeting sequence (MTS) required for the trafficking of IF\(_1\) into the mitochondrial matrix. In complex, IF\(_1\) shows an ordered N-terminal region, which adopts a helix-turn-helix structure (HTH: residues 14–47) necessary for a correct interaction with the F\(_1\) domain of the ATP synthase. A calmodulin-binding site (CBS: residues 33–42) have been identified at positions 33–42, followed by a histidine-rich region (HRR: residues 48–70) which is implicated in the pH-sensing mechanism and hence in the dimerization. The dimerization of IF\(_1\) depends on the C-terminal region, which hosts the dimerization domain (DD: residues 37–84), while the oligomerization domain (OD: residues 32–44) is located in the N-terminal region of the protein, so that after oligomerization the inhibitory domain is hidden and the protein inactivated. (b) Immunocytochemical localization of IF\(_1\) in HeLa cells: the preferential mitochondrial matrix compartmentalization of the protein is shown by its colocalization with the ATP synthase. Cells were costained with anti-IF\(_1\) and anti-F\(_1\)F\(_0\)-ATP synthase \( \beta \) chain antibodies, while DAPI was used for nuclear counterstaining.
by preventing its hydrolysis, when the $H^+$ electrochemical gradient across the mitochondrial inner membrane is lost (e.g., during hypoxic/ischaemic conditions) and the enzyme reverses its activity to transiently restore $\Delta \Psi_m$ [58, 59].

IF$_1$ is encoded by the ATPIF1 gene, localized at chromosome 1, and is synthesized as a propolypeptide (106 residues in humans [52]) harbouring a highly conserved N-terminal presequence of 25 amino acids (identified by the comparison between the amino acid sequences deduced from the cDNA and the purified protein) presumably important for the mitochondrial targeting of the protein [61]. The mature polypeptide (81–84 aa in mammals) is significantly conserved among various species. The human protein exhibits 67–74% of sequence homology with other known mamalian IF$_1$ inhibitors, and also partial homology with the yeast inhibitor Inh1p [52]. Interestingly, there is a strong correlation between the high sequence conservation and function, as IF$_1$ from one species is able to inhibit the F$_i$F$_o$-ATPase synthase from another, including yeast [51, 62, 63]. Conversely, yeast IF$_1$ is not able to inhibit the animal F$_i$ domain because its activity is stabilized by accessory proteins which have no homologues in animals [64].
The most conserved segment of the inhibitory factor is the region that interacts with the F$_1$F$_0$-ATP synthase, comprising amino acid residues 20–50 (90% identity between the human and bovine sequences, 80% between bovine and rodent [61]), and a lot of research into understanding the minimal inhibitory sequence of IF$_1$ has focused on this sequence. Van Raaij and coworkers investigated this [65], by measuring the activity of several truncated forms of the protein. The intact and truncated forms assayed for inhibition of F$_1$F$_0$-ATP synthase using IF$_1$-depleted submitochondrial particles revealed that the minimal inhibitory sequence consists of residues 14–47.

In 2001, Ichikawa and colleagues [66] showed by amino acid replacement that in yeast five residues (F17, R20, R22, E25 and F28) are essential for the inhibitory activity of the protein. Two years later it was brought up that six homologous residues of bovine IF$_1$ (F22, R25, E26, Q27, E30, and Y33), which form a cluster on the surface of the $\alpha$-helix [67], represent the inhibitory site of the protein [68].

5. IF$_1$: Molecular Structure and Conformational Changes

The molecular structure of IF$_1$ was initially characterized by Cabezon et al. in 2001 [67] (2.2 Å resolution X-ray crystallographic analysis). The protein is $\alpha$-helical along most of its length (∼90 Å [69]) and is active as a dimer at low pH (≤6.7). The dimerization of two IF$_1$ monomers involves their C-terminal regions (residues 37–84 [70]), which form an antiparallel double-stranded coiled-coil unit stabilized by complimentary hydrophobic interactions between the two helices, involving residues 49–81 [67]. Within the dimer, the two minimal inhibitory sequences are at the opposite ends (the dimer shows an end-end distance of at least 130 Å, while the distance between the two inhibitory regions is 62 Å) and can react with two F$_1$ domains simultaneously (Figure 3(c)). When the pH is above neutrality (7.0–8.0), the dimers can assemble inactive tetramers or higher oligomers by forming antiparallel coiled-coils in the N-terminal regions (residues 32–44 are involved [70]). Every protomer of the dimer can participate in two coiled-coil units with two different helices, binding two dimers simultaneously. The formation of the oligomers masks the inhibitory sequences of the dimers, so that IF$_1$ cannot bind the F$_1$F$_0$-ATP synthase (Figure 3(a)). Dimers and oligomers are in equilibrium at pH 6.5 [70].

Mammalian IF$_1$ contains five highly conserved histidines (at positions 48, 49, 55, 56 and 70) that, if chemically modified or replaced, lead to a complete loss of the pH-susceptible activity of the protein without affecting its inhibitory capacity [60, 71]. This histidine-rich region (residues 48–70) is involved in the pH sensing mechanism of bovine IF$_1$, and undergoes conformational changes depending on acidity or alkalinity of the environment [61]. Critical for the pH-dependent interconversion between the two aggregation states of the polypeptide is the histidine 49 [70, 71].

It was observed that replacement of this residue with a different amino acid induces full activation of IF$_1$ at pH 8, and abolishes the ability of the dimers to form oligomers [67, 71]. The five histidines seem to be important for the pH-regulated decrease in activity between pH 6.7 and 8.0, even if they may not represent the only mechanism responsible for such regulation. In fact, a pH-dependent activity was also observed in the IF$_1$ 22–46 peptide [72] and detected in a 12-residue segment from 32 to 43 [70]; moreover, H49 is not conserved in yeast, suggesting a diverse pH-sensitivity of the protein [73].

Apart from controlling the oligomerization of the polypeptide and consequently the availability of its inhibitory site, the pH itself was proposed to represent the switch between inactive and active forms by controlling the helical content and the flexibility of the whole protein [74]. At low pH (∼6.7), the helical content seems to be lower, so that the N-terminal region is less ordered and, instead of forming a coiled-coil unit with two other dimers, assumes the correct conformation for binding the F$_1$ complex. However, this theory is controversial since pH was thought to simply act by modulating the electrostatic interactions between the polar residue-enriched N-terminal regions of the dimers, as presented by Cabezon et al. [67]. Recently, Ando and Ichikawa [73] discovered that pH could effectively change the conformation of the active site by acting on a highly conserved glutamate residue, E26 in bovine IF$_1$ or E21 in yeast IF$_1$ (H49 is not conserved in yeast and, as a consequence, cannot represent the only pH sensor residue of the protein). The mechanism of pH-dependency mediated by glutamate regulates only the inhibitory activity of the F$_1$-binding site and not the aggregation state of the polypeptide.

Very little is known about the transcriptional and post-transcriptional regulations of IF$_1$, despite speculations on a possible contribution by the hypoxia-inducible factor 1-α (HIF1-α) [75] and evidence for a downregulation mediated by the Immediate early response gene X-1 (IEX-1) [76].

6. IF$_1$ in Complex with the F$_1$F$_0$-ATP synthase: Outcome on Mito-Ultrastructure

The molecular crystal structure of bovine IF$_1$:F$_1$-ATPase complex is available [69, 77]. In the complex, IF$_1$ shows an ordered N-terminal region and a disordered C-terminal part. The former adopts a helix-turn-helix structure, in which the two $\alpha$-helices extend between residues 14–18 and 21–50 and are linked by a turn from residues 19–20; residues 4–37 are directly involved in binding the F$_1$ domain. Residues 4–18, which are disordered in the dimer, are instead resolvable after binding.

In the bound form, the inhibitory sequences of the two protomers in the dimer become closer (their distance is shortened from 62 Å to 31 Å). This is possibly due to the flexibility of the C-terminal coiled coil region, which probably has a greater curvature in the complex. The long helix at the N-terminal region of IF$_1$ is inserted almost completely into the F$_1$ domain; only residues 47–50 lie outside of it. This points to the central axis of the $\gamma$-subunit and forms a ~45° angle with it, heading from the external surface towards the central cavity.
The contacts between the bovine inhibitor and the inhibited F1 domain are essentially located at the interface between the αDP− and βDP− subunits, even though IF1 also contacts small portions of the γ−, αβ− and βTP− subunits [77]. Although the inhibitory sequence is comprised of residues 14 to 47, this is not the only region that interacts with the F1Fo-ATP synthase. In fact, as demonstrated by the time-dependent loss of inhibition seen by van Raaij and co-workers [65], residues 1–13 and 48–56 are important for stabilizing the structure (the first peptide interacts directly with the F1 domain, while the second probably contributes to stabilize only the IF1 dimer).

With different approaches, Cabezón et al. [77] and Ichikawa et al. [66] found that a cluster of six residues, F22, R25, E26, Q27, E30 and Y33 in bovine IF1, is essential for the inhibitory activity of the peptide. In the crystal structure of the F1:F1 complex, the three core residues of the cluster, F22, E26 and E30, interact with three highly conserved residues of ββ′ (βR408, βR412, and βE454) which are essential for the enzymatic catalysis and regulation of the F1Fo-ATP synthase [78]. E30 directly interacts with the β-subunit of F1 and seems to be essential for the F1-IF1 interaction (Figure 3). Interestingly, one of the three residues, E26, is also implicated in the pH sensing mechanism of IF1. In 2008, Ando and Ichikawa [73] discovered the key role of the glutamate residue in the pH-dependent activity of the inhibitory protein. They proposed that it was the high pH, by inducing the dissociation of the carboxyl group of E26, to affect the conformation or direction of the side chain of the neighboring residue E30, thus destabilizing the interaction between IF1 and the F1 domain.

As proposed by Walker and co-workers [69], IF1 seems to inhibit the hydrolysis of ATP that occurs in mitochondria under hypoxic conditions by initially binding to the αE/βE−interface of the F1 domain, an event that appears to require the presence of ATP in the active site. It was also suggested that the binding of ATP induces a conformational change in the ββ′-subunit, which creates the binding site for IF1 [77].

Garcia and colleagues provided the first compelling evidences for a role of IF1 in promoting the dimerization of the F1Fo-ATP synthase, and for its involvement in the biogenesis of mitochondrial cristae [79]. Moreover, they also showed that, by increasing the expression of IF1 in rat liver or AS-30D hepatoma mitochondria, a rise in the dimer/monomer ratio of the F1Fo-ATP synthase (correlated with an increase in the enzyme activity) is obtained, while the removal of the inhibitory protein from rat liver or bovine heart mitochondria resulted in a reduction of that ratio [8]. The dimerization of the enzyme is essential for a correct biogenesis of mitochondrial cristae; in fact, it represents a prerequisite for the generation of larger oligomers with a ribbon-like structure that promotes curvature and growth of tubular cristae membranes [80].

In a recent study, we demonstrated the pivotal role of IF1 in cell physiology through promotion of the F1Fo-ATP synthase dimerization. Briefly, we showed that IF1 overexpression efficiently increase the activity and the ratio of dimeric to monomeric forms of the F1Fo-ATP synthase, with augmented cristae number, mitochondrial membrane stability, and mitochondrial inner volume [81], thus ensuring a correct mitochondrial inner structure. This is a phenomenon of secure relevance for apoptosis.

7. IF1 in Cell Pathology: from Limitation of the Mitochondrial ATP Consumption to Anaemia

Decreased ΔΨm induces the reversion of the F1Fo-ATP synthase, which starts hydrolysing ATP in the attempt to restore the H+ gradient through the mitochondrial inner membrane, transforming mitochondria in ATP consumers (Figure 3(b)). This condition associates with ischaemia, in which the interruption of tissue blood flow causes a reduction of cell oxygenation (hypoxia) inhibiting mitochondrial respiration. Clear evidence for the reversal of the F1Fo-ATP synthase during ischaemia can be obtained by using an ETC inhibitor (e.g., rotenone, which acts on complex I, or NaN3, which inhibits complex IV) and concomitantly adding oligomycin, an antibiotic that blocks the F1Fo-ATP synthase. These cells will experience a reduced depletion of ATP compared to those bathed with the ETC inhibitor but without oligomycin.

The negative effect of the reversal of the F1Fo-ATP synthase is coupled to the reversal of the adenosine nucleotide translocator (ANT) [82], an IMM transmembrane complex that, in physiological conditions, mediates the exchange of cytosolic ADP and mitochondrial matrix ATP (utilizing the different gradients between the two compartments). Reverse activities of both F1Fo-ATP synthase and ANT transform mitochondria from ATP producers to ATP consumers, leading to massive cytosolic ATP depletion in hypoxic cells following ischaemia.

Reduced intracellular ATP level is flanked by elevated cytoplasmatic H+, Na+, and Ca2+ concentrations, inducing osmotic loading and mitochondrial/endoplasmatic reticulum injury so that, during ischaemia, death of hypoxic cells by necrosis easily occurs.

Early reperfusion minimizes the extent of cellular damage, salvaging cells within ischaemic regions from necrosis, but it can also causes lethal injury to cells with severe ischaemia-induced metabolic derangements (reviewed in [83]). In the latter case, reperfusion alters the activity of plasma membrane transporters (e.g., abolishing acidosis-mediated inhibition of the Na+-Ca2+ exchanger and inducing the activation of calpain, which disturbs Na+-K+ pump function), thus leading to massive influx of Ca2+ into the cytosol and mitochondrial Ca2+ overload. At the same time, resupply of oxygen to mitochondria restores ATP production but also induces a rise in reactive oxygen species (ROS) production. Both mitochondrial Ca2+ overload and augmented ROS levels represent the prelude to the opening of the mitochondrial permeability transition pore (mPTP) and cell death.

The entity of the reversal of the F1Fo-ATP synthase and its action as an H+ motive ATPase during oxygen deprivation were shown by Jennings et al. in 1991 [7]. Studying the changes in ATP depletion and anaerobic glycolysis in totally ischaemic dog heart after inhibiting the F1Fo-ATP synthase
via oligomycin, the same authors have evaluated that about 35% of ATP utilization during the first 90 minutes of total ischaemia in the dog heart is due to the reversion of the enzyme activity.

Oligomycin-mediated inhibition slows down ATP depletion during ischaemia. Rouslin and colleagues [11] proved that the antibiotic has a very small and transient effect on mitochondrial function when used in fast heart-rate animals, like rats, if compared to slow heart-rate species, like larger mammals are. The same authors have subsequently shown that this diversity depends on the different F$_1$F$_0$-ATP synthase: F$_1$ ratios, with a diverse ability to inhibit mitochondrial-driven consumption of ATP when needed [84].

The estimation was that F$_1$ reduces the ATP-hydrolysing activity of F$_1$F$_0$-ATP synthase during ischaemic conditions by up to 70–80% (in slow heart-rate species) [85], thus preventing cellular damage due to ischaemic conditions and delaying cell death when oxygen and glucose are limited. Upon reperfusion, the binding of F$_1$ to the F$_1$F$_0$-ATP synthase is quickly reversed [86], so that sublethal ischaemic episodes could be followed by a relatively rapid recovery of intracellular ATP.

Although this model is challenged by later evidence showing that in rat heart, during ischaemic preconditioning, mitochondrial ATP hydrolysis is inhibited probably as a consequence of the binding of F$_1$ [87], the variations in ratio between the enzyme and its controller among animal species are still a fascinating possibility. Nonetheless, this ratio differs per se among organs and cell types of the same organ [12].

By modulating the expression of F$_1$ in human (HeLa) and murine (C2C12) cells, we demonstrated that, when F$_1$ is overexpressed, cells show a decrease in ATP consumption [81]. Thus, variations in F$_1$ expression could influence cellular or tissue resistance to ischaemic injury in different species or cell types.

### 7.1. Central Nervous System

Notably, F$_1$ expression is elevated in highly oxidative cells, like neurons and kidney proximal tubules [12], which are highly susceptible to mitochondria deregulations. In the central nervous system, for example, neurons and astrocytes show a great difference in the F$_1$ : F$_1$-β-subunit ratio, which is ~1.45 in the former and ~0.8 in the latter; as a consequence, the inhibition of respiration with NaN$_3$ causes a progressive loss of ΔΨ$_m$ in neurons, while in astrocytes the proton gradient is maintained at a new steady state [81]. Thus, higher levels of F$_1$ could be advantageous in cells highly depending on oxidative phosphorylation by preventing ATP depletion and quick cellular damage during ischaemia.

### 7.2. Preconditioning

A final interesting aspect is the highly probable involvement of F$_1$ in the ischaemic preconditioning mechanism. This phenomenon, which is characterized by the acquisition of a strong resistance to ischaemia in tissue undergoing brief, repeated periods of sublethal ischaemia, is commonly observed in heart, skeletal muscle and brain [61]. It is described as a slowing of energy metabolism with a decreased rate of ATP depletion during ischaemia [88]. F$_1$ is proposed to take part in this process after the observation that rat heart preconditioning associates with the inhibition of mitochondrial ATP hydrolysis during ischaemia [87]. This was later confirmed by Penna et al., who showed that decrease in the enzyme’s activity after ischemic preconditioning correlates with an augmented binding of F$_1$ [89]. Moreover, a consensus exists that the opening of mitochondrial ATP-sensitive K$^+$ channels plays a central role in cell protection during ischemic preconditioning, causing mitochondrial deenergisation and acidification owing to H$^+$/K$^+$ exchange [86]; both effects are likely to promote the binding of F$_1$ to the ATP synthase.

### 7.3. Cancer

The importance of mitochondrial metabolism in cancer cells is underlined by the frequently observed, close interaction of glycolytic enzymes with mitochondria. This creates a mutually sustaining relationship between glycolysis, which represents the primary metabolic pathway for tumours sustenance [90], and oxidative phosphorylation.

Regarding the F$_1$F$_0$-ATP synthase endogenous regulator, F$_1$, its overexpression has been observed in many human carcinomas (including lung, colon, breast, and cervix carcinomas [10], Ehrlich ascites carcinoma [91], Zajdela hepatoma and Yoshida sarcoma [92]), but little is still known about the associated effects, and the few theories that have been put forward are highly controversial. Increased expression of the protein is associated with a higher binding efficiency to the F$_1$F$_0$-ATP synthase [93], suggesting a greater protection of cancer cells against energy dissipation upon F$_1$F$_0$-ATP synthase reversal. This was theorized by Chernyak et al. [91], although more recent work has cast doubt on this hypothesis, revealing a relationship between the inhibitor overexpression in human carcinomas and an increase in both ΔΨ$_m$ and glycolytic rate [10].

The protein may therefore be involved in protecting tumour cells from cytosolic ATP depletion and excessive reactive oxygen species (ROS) production (the majority of tumours have little or no vascularization, so that cancerous cells grow in a hypoxic environment). Over and above that, to guarantee cell viability, mitochondria should not become ATP consumers. ATP depletion, ROS imbalance, low cytosolic pH and oxidation of NAD(P)H facilitate the opening of the mitochondrial permeability transition pore (mPTP) and the activation of the intrinsic apoptotic pathway [94]. Numerous studies have also demonstrated that transient hyperpolarization of the mitochondrial membrane can lead to cell apoptosis [38, 95]. By preventing ROS production and IMM hyperpolarization, F$_1$ could also protect cancer cells from ROS-mediated apoptosis. F$_1$ induces the dimerization of the F$_1$F$_0$-ATP synthase [8], which might play an essential role in preventing both mitochondrial network fragmentation and cytochrome c release from mitochondrial cristae, thus inhibiting the activation of the intrinsic apoptotic pathway. Our previous studies seem to support these hypotheses [81], and, future and focused studies will shed light on this (Tan et al., under review). Moreover, an original recent work by Cuezva and co-workers has elegantly demonstrated that F$_1$ is protective against chemotherapy.
and supports cell proliferation of cancerous cells via the NfkB pathway [96]. Although this work focuses principally on the postmitochondrial effects of IF1, it is anyway a compelling evidence for a contribution to neoplastic degeneration and resistance to apoptosis. A starting point to unravel how mitochondrial structure and function are primed by IF1 overexpression, and to understand to what extent this dictates cellular transformation.

7.4. Luft’s Disease. To date, the absence of IF1 has been correlated with only one human pathological condition of neuronal origin: a mitochondrial myopathy called Luft’s disease, characterized by nonthyroidal hypermetabolism and densely packed mitochondrial cristae (it is one of the rarest of the mitochondrial diseases, with only two reported cases). Basal ATPase activity in one of the patients was seven times higher than normal [13], and no IF1 activity was detected in fibroblasts cultured from the skeletal muscle [14]; however, no mutations in the ATPF1 gene were identified, and the genetic cause of the disease remains obscure.

7.5. Anaemia. Despite what discussed above, we have very recently collected evidences for a deficiency in the ATPF1 gene associated with a form of hypochromic anaemia (Shah et al., under review). The mechanism we propose is related with the secondary effect that the absence of IF1, and the consequent lack in its inhibitory activity, on the F1F0-ATP synthase reversion, has on $\Delta \Psi_m$ and matrix pH. It is known that erythroids’ differentiation is triggered by a decrease in $\Delta \Psi_m$, that is responsible for a critical redistribution of intracellular Ca$^{2+}$ and a transient activation of caspases [97]. Anyway, we found that the increase in mitochondrial matrix pH, which is observed in zebrafish models and murine cells carrying the mutated form of the ATPF1 gene, is causally linked to a decrease in ferrochelatase activity, which leads to defects in the incorporation of $^{59}$Fe into protoporphyrin IX to generate the hemoglobin prosthetic group heme.

Such a remarkable finding puts IF1 amongst the regulators of heme biosynthesis, not only describing a new mechanism for sideroblastic anaemia, but also confirming the involvement of the inhibitory protein in human pathologies related to mitochondrial disorders.

8. Conclusions

The F1F0-ATP synthase is a wonderful machinery, with the unique capacity of producing and consuming energy, if necessary, to preserve the integrity of the organelle to which it belongs. A precise and sustainable way to regulate its activity is therefore paramount, and the Inhibitory Factor 1, a protein encoded by the nuclear DNA, represents the molecule deputed to do so. In the face of a well-defined biochemistry, its role in cell physiology and mitochondrial anatomy has been only recently discovered, posing the protein at the crossroad between dynamics and energy balance. This, together with growing evidence for a contribution to cell and tissue pathology, leads to novel ways to investigate and thoroughly address IF1 functional biology.

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