Supporting Information

for

Ultrasound-assisted Strecker synthesis of novel 2-(hetero)aryl-2-(arylamino)acetonitrile derivatives

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Experimental

1. Materials and apparatus

The starting Schiff bases 1a–j were prepared by the microwave-assisted condensation of the corresponding (hetero)aromatic aldehydes with aromatic primary amines according to our previously reported procedure [1]. Commercial reagent grade TMSC and the PEG solvent were purchased from Sigma Aldrich and employed without further purification. Thin layer chromatography was performed on Merck DC Alufolien, silica gel 60 F$_{254}$, and components were visualized by UV VL-4LC. The melting points are uncorrected and were determined in capillaries with an Electrothermal 9100 instrument. All reactions were carried out in an ultrasonic bath Elmsasonic S 15 (H), Elma Schmidbauer GmbH, Gottlieb-Daimler-Str. 17, D-78224 Singen, Germany.

2. Spectral measurements

HRMS spectra were recorded using a Thermo LTQ Orbitrap XL instrument. NMR spectra were recorded at room temperature on a 600 MHz Bruker Avance instrument. Chemical shifts are expressed in terms of δ (ppm), relative to the standard tetramethylsilane (TMS). FTIR spectra were recorded using an ATR Bruker Vector 22 instrument. A digital Polarimeter, Perkin Elmer 341 was used for the measurement of the optical rotation angle. The experiment was conducted at room temperature using the electromagnetic radiation with λ=589.3 nm (sodium D-line). The sample contained a solution (0.5 mg/mL) of α-(arylamino)acetanitride 2a in ethanol, placed in a 100 mm tube. The SEM measurements were performed using a JEOL JSM 5600 LV microscope, equipped with an EDX spectrometer, Oxford Instruments (INCA 200 software). The energy of the acceleration beam employed was 15 kV and the given results are 200× magnifications.
crystallographic data were collected on a Bruker SMART APEX diffractometer by using graphite-monochromatic MoK\(\alpha\) radiation (\(\lambda = 0.71073\) Å) at room temperature (294 K). The structures were refined with anisotropic thermal parameters and the hydrogen atoms were refined with a riding model and a mutual isotropic thermal parameter. For structure solving and refinement the software package SHELX-97 was used [2]. The drawings were created with the Diamond program [3]. Crystallographic data (excluding structure factors) for the structure 2a in this paper have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication no. 2018198 CCDC. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK, (fax: +44-(0)1223-336033 or by email: deposit@ccdc.cam.ac.uk).

3. General procedure for the preparation of \(\alpha\)-arylamino-acetonitrile derivatives

\begin{align*}
\text{a) Ultrasound-assisted reaction conditions} \\
\text{The reaction mixture prepared by adding TMSCN (1 equiv) to a solution containing the aldimine (1 equiv) dissolved in PEG (5 mL) and water (1 mL), was placed in a 25 mL beaker and sonicated (37 kHz, 95 W) for 30 minutes at 25 °C. After completion of the reaction, the product was collected by filtration directly from the reaction mixture, or after being poured into water. The crystalline product collected by filtration was dried and, if required, further purification can be performed by recrystallization.} \\
\text{b) Classical conditions} \\
\text{The reaction mixture, prepared according to the procedure described above (a), was stirred at room temperature for 3 days. After completion of the reaction, the mixture was poured into water and the product was extracted in diethyl ether. After evaporation of the organic solvent, the solid product was purified by recrystallization.}
\end{align*}

2-(10-Methyl-10\(H\)-phenothiazin-3-yl)-2-(p-tolylamino)acetonitrile (2a)

Recrystallization from isopropanol gave a light yellow solid, yield: 95 % (0.30 g); m.p. 94-95 °C; FT-IR (KBr, \(\tilde{\nu}_{\text{max}}/\text{cm}^{-1}\)) 3357 (\(\nu_{\text{N-H}}\), m), 2991, 2959, 2835 (\(\nu_{\text{C-H}}\), m), 2231 (\(\nu_{\text{C≡N}}\), w). HRMS (ESI): m/z calcd. for C\(_{22}\)H\(_{19}\)N\(_3\)S [M\(^+\)], 357.1294; found 357.1295; \(^1\)H NMR (600 MHz, CDCl\(_3\)) \(\delta\) ppm: 2.30 (s, 3H, CH\(_3\)), 3.40 (s, 3H, N-CH\(_3\)), 3.89 (br, 1H, NH), 5.31 (s, 1H, CHCN), 6.98 (d,
2H, J=8.4Hz, H_{Ph2'}), 6.83-6.86 (m, 2H, H1, H9), 6.99 (td, 1H, J=7.5Hz, J=0.9Hz, H7), 7.10 (d, 2H, J=8.4Hz, H_{Ph3'}), 7.16 (dd, 1H, J=7.6Hz, J=1.3Hz, H6), 7.22 (td, 1H, J=7.8Hz, J=1.3Hz, H8), 7.36 (d, 1H, J=1.9Hz, H4), 7.39 (dd, 1H, J=8.4Hz, J=1.9, H2); 13C NMR (150 MHz, CDCl3) δ ppm: 20.5 (CH3), 35.4 (NCH3), 49.9 (CH), 114.2 (C1), 114.3 (C9), 114.5 (2C2'), 118.3 (q CN), 122.6 (q C4a), 122.9 (C7), 124.7 (q C5a), 125.8 (C6), 126.8 (C8), 127.2 (C4), 127.7 (C2), 128.0 (q C4'), 129.7 (q C3), 130 (2C3'), 142.3 (q C1'), 145.1(q C10a), 146.8(q C9a).

2-(4-Methoxyphenylamino)-2-(10-methyl-10H-phenothiazin-3-yl)acetonitrile (2b)

Recrystallization from ethyl acetate gave a yellow solid, yield: 95 % (0.30g); m.p. 144-145 °C; FT-IR (KBr, \( \tilde{\nu}_{max}/cm^{-1} \)) 3350 (\( \nu_{N-H} \), m) 2960, 2832 (\( \nu_{C-H} \), m), 2233 (\( \nu_{C=\bar{N}} \), w), 1033 (\( \nu_{C-O} \), m). HRMS (APCI): m/z calcd. for C21H19N2SO [M-CN]+, 347.1218; found [M-CN]+ 347.1236. 1H NMR (600 MHz, CDCl3) δ ppm: 3.40 (s, 3H, N-CH3), 3.76 (br, 1H, NH), 3.79 (s, 3H, OCH3), 5.26 (s, 1H, CNCH), 6.76 (d, 2H, J=8.8Hz, H_{Ph2'}), 6.83-6.87 (m, 4H, H_{Ph3'}, H1, H9), 6.99 (td, 1H, J=7.5Hz, J=1.0Hz, H7), 7.17 (dd, 1H, J=7.5Hz, J=1.2Hz, H6), 7.23 (td, 1H, J=7.5Hz, J=1.3Hz, H8), 7.36 (d, 1H, J=1.9Hz, H4), 7.39 (dd, 1H, J=8.4Hz, J=1.9, H2); 13C NMR (150 MHz, CDCl3) δ ppm: 35.4 (NCH3), 50.7 (CH), 55.6 (OCH3), 114.2 (C1), 114.3 (C9), 115.0 (2C2'), 116.3 (2C2'), 118.4 (q CN), 122.6 (q C4a), 122.9 (C7), 124.7 (q C5a), 125.8 (C6), 126.4 (C8), 127.2 (C4), 127.7 (C2), 128.1 (q C3), 138.5 (q C1'), 145.1 (q C10a), 146.8 (q C9a), 154.1 (q C4').

Methyl 4-((cyano(10-methyl-10H-phenothiazin-3-yl)methyl)amino)benzoate (2c)

Recrystallization from ethanol gave an orange solid, yield: 91 % (0.29g); m.p. 253 °C decomp.; FT-IR (KBr, \( \tilde{\nu}_{max}/cm^{-1} \)) 3340 (\( \nu_{N-H} \), w), 2975, 2870, 2135 (\( \nu_{C=N} \), w), 1680 (\( \nu_{C=O} \), s). HRMS (APCI): m/z calcd. for C25H21N2SO2 [M-CN]+, 375.1167; found [M-CN]+ 375.1170. 1H NMR (600 MHz, CDCl3) δ ppm: 3.20 (s, 3H, N-CH3), 3.81 (s, 3H, O-CH3), 4.83 (br, 1H, NH), 5.38 (d, 1H, J=7.7Hz, CNCH), 6.68 (d, 2H, J=8.4Hz, H_{Ph2'}), 6.74 (d, 1H, J=8.2, H1), 6.78 (d, 1H, J=8.1, H9), 6.91 (t, 1H, J=7.5Hz, H7), 7.07 (d, 1H, J=7.5Hz, H6), 7.14(t, 1H, J=7.5Hz, H8), 7.27-7.28 (m, 2H, H2, H4), 7.85(d, 2H, J=8.4Hz, H_{Ph3'}). 13C NMR (150 MHz, CDCl3) δ ppm: 35.3 (NCH3), 48.3 (CH), 51.7 (OCH3), 112.9 (2 C2'), 114.3 (C1), 114.4 (C9), 117.7 (q CN), 120.8 (qC4'), 122.3 (q C4a), 122.9 (C7), 124.6 (q C5a), 125.6 (C6), 126.4 (C8), 127.1 (C4), 128.8 (C2, q C3), 131.8 (2C3'), 145.0 (q C10a), 146.7 (q C9a), 148.7 (q C1'), 167.0 (q C=O).
4-((Cyano(10-methyl-10H-phenothiazin-3-yl)ethyl)amino)benzoic acid (2d)

Recrystallization from isopropanol gave a light brown solid, yield: 97% (0.31g); m.p. 172-174 °C; FT-IR (KBr, $\tilde{\nu}_{\text{max}}$/cm$^{-1}$) 3440 (\nu_{\text{NH}}, w), 3081 (\nu_{\text{CH}}, broad), 2962, 2867, 2874, 2242 (\nu_{\text{C=O}}, w), 1696 (\nu_{\text{C=O}}, s); HRMS (ESI): m/z calcd. for C$_{22}$H$_{18}$N$_{3}$SO$_{2}$ [M+H] $^+$ 388.1120; found 388.2554; $^1$H NMR (600 MHz, CD$_2$COCD$_3$) $\delta$ ppm: 3.40 (s, 3H, CH$_3$), 5.91 (s, 1H, CNCH), 6.6 (br, 1H, NH), 6.93-7.03 (m, 5H, H$_7$, H$_{Ph2'}$), 7.16(d, 1H, J=7.2Hz, H$_6$), 7.23(t, 1H, J=7.4Hz, H$_8$), 7.39 (s, 1H, H$_4$), 7.48(d, 1H J=7.8Hz, H$_2$) 7.91(d, 2H, J=8.2Hz, H$_{Ph3'}$). $^{13}$C NMR (150 MHz, CDCl$_3$) $\delta$ ppm: 34.9 (NCH$_3$), 47.7(CH) 112.8(2C$_2$-), 114.6(C$_1$, C$_9$), 118.3 (q CN), 120.5 (q C$_4$), 122.2 (q C$_{4a}$), 122.8 (C$_7$), 123.9 (q C$_{5a}$), 125.6 (C$_6$), 126.8 (C$_8$), 126.9(C$_4$), 127.9 (C$_2$), 128.5 (q C$_3$), 131.4(2C$_3'$), 145.3 (q C$_{10a}$), 146.6 (q C$_{9a}$), 148.7 (q C$_1'$), 167.0 (q C=O).

2-(4-Chlorophenylamino)-2-(10-methyl-10H-phenothiazin-3-yl)acetonitrile (2e)

Recrystallization from ethyl acetate gave a light yellow solid, yield: 93% (0.29g); m.p. 206-207 °C; FT-IR (KBr, $\tilde{\nu}_{\text{max}}$/cm$^{-1}$) 3335 (\nu_{\text{NH}}, w), 2969, 2873, 2228 (\nu_{\text{C=O}}, w). HRMS (ESI): m/z calcd. for C$_{20}$H$_{16}$ClN$_2$S [M-CN]$^+$ 351.0717; found 351.0741; $^1$H NMR (600 MHz, CDCl$_3$) $\delta$ ppm: 3.41 (s, 3H, CH$_3$), 5.21 (q CH, 4.03(d, 1H, J=8.0Hz, NH), 5.85 (d, 1H, CNCH, =8.0Hz), 6.30 (d, 2H, J =8.8Hz, H$_{Ph2'}$), 6.84-6.87 (m, 2H, H$_1$, H$_9$), 6.93 (dt, 1H, J=7.6Hz, J=0.8Hz, H$_7$), 7.17 (dd, H, J=7.6Hz, H$_6$), 7.21–7.24(m, 2H, H$_8$, H$_{Ph3'}$), 7.34 (d, 1H, J=2.1Hz, H$_3$), 7.38 (dd, 1H, J=8.3Hz, J=2.1, H$_2$); $^{13}$C NMR (150 MHz, CDCl$_3$) $\delta$ ppm: 35.4 (NCH$_3$), 49.5 (CH), 114.33 (C$_1$), 114.39 (C$_9$), 115.3(2C$_2$), 117.8 (q CN), 122.5 q C$_{4a}$, 123.0 (C$_7$), 125.0 (qC$_{5a}$), 125.1 (q C$_4$), 125.7 (C$_6$), 126.3 (C$_8$), 127.2 (C$_4$), 127.3 (q C$_3$), 127.8 (C$_2$), 129.4(2C$_3'$), 143.1 (q C$_{10a}$), 145.0 (q C$_{10a}$), 147.0 (q C$_1'$).

2-(10-Methyl-10H-phenothiazin-3-yl)-2-(4-nitrophenylamino)acetonitrile (2f)

Purified by recrystallization from isopropanol, then from ethyl acetate gave a yellow-brown solid, yield: 96% (0.30g); m.p. 151-152 °C; FT-IR (KBr, $\tilde{\nu}_{\text{max}}$/cm$^{-1}$) 3378 (\nu_{\text{NH}}, w), 2956, 2819, 2238 (\nu_{\text{C=O}}, w), 1531, 1352 (\nu_{\text{asim NO}}, s). HRMS (APCI): m/z calcd. for C$_{20}$H$_{16}$N$_{4}$SO$_{2}$ [M-CN]$^+$, 362.0963; found [M-CN]$^+$ 362.0964. $^1$H NMR (600 MHz, CDCl$_3$) $\delta$ ppm: 3.31(s, 3H, NCH$_3$), 5.07 (br, 1H, NH), 5.45(1H, d, J=7.2Hz, CH), 6.69 (d, 2H, J =9.1Hz, H$_{Ph2'}$), 6.77 (d, 1, J=8.1Hz, H$_1$), 6.78 (d, 1, J=8.4Hz, H$_9$), 6.91 (t, 1H, J=7.5Hz, H$_7$), 7.07(dd, 1H, J=7.5Hz, J=1.3Hz, H$_6$), 7.14 (td, 1H, J=8.4 Hz, J=1.3HZ, H$_8$), 7.25 (d, 1H, J=2.0Hz, H$_4$), 7.30 (dd, 1H, J=8.1Hz, J=2.0, H$_2$), 8.04(d, 2H, J =9.1Hz, H$_{Ph3'}$). $^{13}$C NMR (150 MHz, CDCl$_3$) $\delta$ ppm: 35.4 (NCH$_3$), 48.1 (CH), 112.7 (2C$_2'$), 114.2 (C$_1$), 114.4 (C$_9$), 117.2 (qCN), 122.3 (qC$_{4a}$), 123.0 (C$_7$), 124.8 (qC$_{5a}$), 125.6
6 (C₆), 126.0(2C₃), 126.4 (C₈), 126.5 (q C₃), 127.1 (C₂), 127.8 (C₄), 139.7 (q C₄'), 144.9 (q C₉a), 149.9 (q C₁₀a), 150.3 (q C₁').

2-(10-Methyl-10H-phenothiazin-3-yl)-2-(3-nitrophenylamino)acetonitrile (2g)
Recrystallization from isopropanol, then from ethyl acetate gave a brown solid, yield: 98 % (0.31g); m.p. 138-139 °C; FT-IR (KBr, δmax/cm⁻¹) 3378 (νN-H, m), 2956, 2819, 2238 (νC=O, w), 1531, 1352 (νas, sim NO, s); HRMS (APCI): m/z calcd. for C₂₀H₁₆N₄SO₂ [M-CN]⁺, 362.0963; found [M-CN]⁺ 362.0966; ¹H NMR (400 MHz, CDCl₃), δ ppm: 3.40 (s, 3H, CH₃), 4.47 (d, 1H, J=7.8Hz, NH), 5.40 (d, 1H, J=7.8Hz, CH), 6.85 (d, 2H, J=8.4Hz, H₁, H₉), 6.97-7.04 (m, 2H, H₆', H₇), 7.15 (dd, 1H, J=7.5Hz, J=1.1Hz, H₆), 7.22 (t, 1H, J=7.5Hz, J=1.1Hz, H₈), 7.59 (d, 1H, J=1.8Hz, H₄), 7.37-7.43 (m, 2H, H₂, H₅'), 7.59 (s, 1H, H₆'), 7.72 (d, 1H, J=8.0Hz, J=1.1Hz, H₉'), 13C NMR (100 MHz, CDCl₃), δppm: 35.4 (NCH₃), 49.0 (CH), 108.3 (C₂'), 114.42 (C₁), 114.45 (C₆), 114.8 (a'), 117.3(CN), 119.6 (C₅'), 122.4 (qC₄a), 123.1(C₇), 125.1 (qC₉a), 125.7 (C₆), 126.4 (C₈), 126.6 (q C₉), 127.2 (C₂), 127.8 (C₄), 130.0 (C₅'), 144.9 (q C₉a), 145.3 (q C₁₀a), 147.2 (q C₁'), 149.2 (q C₃').

2-(4-Methoxyphenylamino)-2-(ferroceny)acetonitrile (2h)
Recrystallization from ethanol gave a brown solid, yield: 90 % (0.29g); m.p. 107-108 °C; FT-IR (KBr, δmax/cm⁻¹) 3328 (νN-H, w) 3103 (νO-H, m), 2955, 2829, 2225 (νC=O, w), 1510, 1106 (νC=O, s), 1463 (νC-O, m); HRMS (APCI): m/z calcd. for C₁₅H₁₈Fe NO [M-CN]⁺, 320.0738; found [M-CN]⁺ 320.1636; ¹H NMR (600 MHz, CDCl₃) δ ppm 3.80 (s, 3H, OCH₃), 3.85(br, 1H, NH), 4.31-4.33 (m, 7H, H-Cp), 4.43 (s , 1H, H-Cp), 4.49 (s , 1H, H-Cp), 5.06 (s, 1H, CNCH), 6.78-88 (m, 4H, H₉'), 13C NMR (150 MHz, CDCl₃) δ ppm: 47.8 (CH), 55.7 (OCH₃), 66.8 (C₇), 68.2(C₉), 68.9(C₅), 69.1(C₇), 69.3(5C overlap), 82.6 (q C₇) , 115.0(2Ph₃'), 116.2(2Ph₂'), 118.6 (CN), 138.7 (q C₁'), 151.0 (q C₄').

Methyl 4-(cyano(ferrocenyl)methylamino)benzoate (2i)
Recrystallization from ethanol gave a light brown solid, m.p. 140-141 °C, yield: 94 % (0.30g); FT-IR (KBr, δmax/cm⁻¹) 3329 (νN-H, s), 2946, 2905, 2869, 2246 (νC=O, w), 1669 (νC=O, s), 1523, 1087 (νC=O, s), 1350 (νC-O, m); HRMS (APCI): m/z calcd. for C₁₉H₁₆FeNO₂ [M-CN]⁺, 348.0687; found [M-CN]⁺ 348.0697; ¹H NMR (400 MHz, DMSO-d₆), δ (ppm): 3.88(s, 3H, COOCH₃), 4.31-4.33(m, 7H, H-Cp), 4.42(s, 6H, H-Cp) 4.47 (s, 1H, H-Cp), 4.69 (d, 1H, J=7.9Hz, NH), 5.19 (d, 1H, J=7.9Hz, CNCH), 6.74 (d, 2H, J=8.6Hz, H₆'), 7.97 (d, 2H, J=8.6Hz, H₈'), 13C NMR (100 MHz, DMSO-d₆), δ ppm: 45.6 (CH), 51.8(OCH₃), 66.8 (C₇), 68.3 (C₇), 69.2 (C₇), 69.4,
69.5 (5C\textsubscript{Cp}, overlap), 81.5 (q C\textsubscript{Cp}), 112.6 (2C\textsubscript{2'}), 117.6 (q CN), 121.1 (q C\textsubscript{4'}), 131.7(2C\textsubscript{3'}), 148.4 (q C\textsubscript{1'}), 166.9 (q CO).

2-(4-Chlorophenylamino)-2-(ferrocenyl)acetonitrile (2j)
Recrystallization from ethanol gave a brown solid, m.p. 107-108°C, yield 92% (0.3g); FT-IR (KBr, \(\bar{\nu}_{\text{max}}/\text{cm}^{-1}\)) 3313 (\(\nu_{\text{NH}, s}\), s), 2915, 2869 (\(\nu_{\text{C-H}, m}\), m), 2239 (\(\nu_{\text{C=N}, w}\), w), 1509, 1077 (\(\nu_{\text{C=C, s}}\)); HRMS (ESI): m/z calcd. for C\textsubscript{17}H\textsubscript{15}ClFeN [M-CN\textsuperscript{+}]\textsuperscript{+}, 324.0242; found [M-CN\textsuperscript{+}]\textsuperscript{+} 324.0363; \(\textsuperscript{1}H\) NMR (600 MHz, CDCl\textsubscript{3}): \(\delta\) ppm 4.18 (br, 1H, NH), 4.46(s, 1H, H-Cp), 4.41(s, 1H, H-Cp) 4.32 (s, 7H, H-Cp), 5.09 (d, 1H, J=6.4Hz, CNCH), 6.70-6.71 (m, 2H, H\textsubscript{Ph2}), 7.23-7.24 (m, 2H, H\textsubscript{Ph3}); \(\textsuperscript{13}C\) NMR (150 MHz, CDCl\textsubscript{3}), \(\delta\) ppm: 46.5 (CH), 66.8(C\textsubscript{Cp}), 68.3(C\textsubscript{Cp}), 69.1(C\textsubscript{Cp}), 69.3(C\textsubscript{Cp}), 69.4 (5C\textsubscript{Cp}, overlap), 81.9 (q\textsubscript{C\textsubscript{Cp}}), 115.1(2C\textsubscript{2'}), 118.1 (q CN), 124.7 (q C\textsubscript{4'}), 129.4 (2C\textsubscript{3'}), 142.3 (q C\textsubscript{1'}).

4-((Cyano(ferrocenyl)methylamino)benzoic acid (2k)
Recrystallization from ethanol gave a dark brown solid, m.p. 177-178°C, yield 94%, (0.3g); FT-IR (KBr, \(\bar{\nu}_{\text{max}}/\text{cm}^{-1}\)) 3312 (\(\nu_{\text{NH}, m}\), m), 3086 (\(\nu_{\text{C-H}, broad}\), broad), 2964 (\(\nu_{\text{C-H}, m}\), m), 2239 (\(\nu_{\text{C=N}, w}\), w), 1674 (\(\nu_{\text{C=O}, s}\), s), 1532, (\(\nu_{\text{C=C}, s}\)); HRMS (ESI): m/z calcd. for C\textsubscript{19}H\textsubscript{16}FeN\textsubscript{2}O\textsubscript{2} 360.0555[M\textsuperscript{+}]; found [M-CN\textsuperscript{+}]\textsuperscript{+} 360.05803; \(\textsuperscript{1}H\) NMR (600 MHz, DMSO-d\textsubscript{6}): \(\delta\) ppm 4.27(s, 1H, H-Cp), 4.30(s, 6H, H-Cp) 4.44 (s, 1H, H-Cp), 4.48(s, 1H, H-Cp ) 5.71 (d, 1H, J=8.5Hz, CNCH), 6.92 (d, 2H, J=8.7Hz, H\textsubscript{Ph2}, H\textsubscript{Ph3}), 6.99 (d, 1H, J=8.5Hz, NH), 7.81 (d, 2H, J=8.7Hz, H\textsubscript{Ph3}), 12.32(1H, COOH); \(\textsuperscript{13}C\) NMR (150 MHz, DMSO-d\textsubscript{6}), \(\delta\) ppm: 45.1 (CH), 68.1 (C\textsubscript{Cp}), 68.9 (C\textsubscript{Cp}), 69.0 (C\textsubscript{Cp}), 69.2 (C\textsubscript{Cp}), 69.7(5C\textsubscript{Cp}, overlap), 82.0 (q C\textsubscript{Cp}), 112.8(2C\textsubscript{2'}), 119.47(q CN), 121.1 (q C\textsubscript{4'}), 131.5(2C\textsubscript{3'}), 150.1 (q C\textsubscript{1'}), 167.8 (q C=O).

Methyl 4-((cyano(phenyl)methylamino)benzoate (2l)
Recrystallization from ethanol gave a yellow solid, yield: 94 % (0.31g); m.p. 128-129 0°C; FT-IR (KBr, \(\bar{\nu}_{\text{max}}/\text{cm}^{-1}\)) 3341 (\(\nu_{\text{NH}, m}\), m), 2251(\(\nu_{\text{C=N}, w}\), w), 1730 (\(\nu_{\text{C=O}, s}\), s), 1120 (\(\nu_{\text{C-O}, s}\), s). MS (EI, 70 eV), m/z (%): 266[M\textsuperscript{+}(100); \(\textsuperscript{1}H\) NMR (400 MHz, CDCl\textsubscript{3}) \(\delta\) ppm: 3.88 (s, 3H, CH\textsubscript{3}), 4.62 (d, 1H, J=7.8Hz, NH), 5.51 (d, 1H, J=7.8Hz, CH), 6.77 (d, 2H, J=8.7Hz H\textsubscript{Ph2}'), 7.49-7.50 (m, 3H, H\textsubscript{Ph3,4}), 7.59-7.61 (m, 2H, H\textsubscript{Ph5}), 7.97(d, 2H, J=8.7Hz, H\textsubscript{Ph3}'); \(\textsuperscript{13}C\) NMR (100.5 MHz, CDCl\textsubscript{3}) \(\delta\)(ppm): 49.4 (OCH\textsubscript{3}), 51.8 (CNCH), 112.9(2C\textsubscript{2'}), 117.6 (q CN), 121.4 (q C\textsubscript{4'}), 127.2(2C\textsubscript{3}), 129.5(2C\textsubscript{2}), 129.8 (C\textsubscript{4}), 131.6(2C\textsubscript{3'}), 133.1 (q C\textsubscript{1}), 148.4 (qC\textsubscript{1'}), 166.9 (q C=O).
4. Crystallographic data

The details of the crystal structure determination and refinement for compound 2a are given in Table S1.

**Table S1:** Crystallographic data for 2-phenothiazinyl-2-(p-tolylamino)acetonitrile (2a).

| Property                              | Details                                      |
|---------------------------------------|----------------------------------------------|
| Empirical formula                     | C$_{22}$H$_{19}$N$_3$S                       |
| Formula weight                        | 357.46 g/mol                                 |
| Temperature                           | 294(2) K                                     |
| Wavelength                            | 0.71073 Å                                    |
| Crystal system                        | Orthorhombic                                 |
| Space group                           | Pca21                                        |
| a [Å]                                 | 13.948(6)                                    |
| b [Å]                                 | 17.457(7)                                    |
| c [Å]                                 | 7.755(3)                                     |
| α [°]                                 | 90                                           |
| β [°]                                 | 90                                           |
| γ [°]                                 | 90                                           |
| Volume                                | 1888.3(13) Å$^3$                             |
| Z                                      | 4                                            |
| Density (calculated)                  | 1.257 g/cm$^3$                               |
| Absorption coefficient                | 0.181 mm$^{-1}$                              |
| F(000)                                | 752                                          |
| Crystal size                          | 0.370 x 0.280 x 0.220 mm                     |
| Theta range for data collection       | 2.333 to 24.987                              |
| Index ranges                          | -16$\leq$h$\leq$16, -20$\leq$k$\leq$20, -9$\leq$l$\leq$9 |
| Reflections collected                 | 17118                                        |
| Independent reflections               | 3315 [R(int) = 0.1230]                       |
| Completeness to theta =24.99°         | 99.8 %                                       |
| Absorption correction                 | Semi-empirical from equivalents              |
| Max. and min. transmission            | 0.961 and 0.936                              |
| Refinement method                     | Full-matrix least-squares on F$^2$           |
| Data / restraints / parameters        | 3315 / 1 / 241                               |
| Goodness-of-fit on F$^2$ -S            | 1.041                                        |
| Final R indices [I>2sigma(I)]         | R1 = 0.0741, wR2 = 0.1451                     |
| R indices (all data)                  | R1 = 0.1073, wR2 = 0.1590                     |
| Largest diff. peak and hole           | 0.334 and -0.235 e/Å$^3$                     |
5. Biological assay
The plate incorporation method [4] was employed for the assessment of the mutagenic and antimutagenic activity for a series of three representatives of the newly synthetized compounds (2c, 2i, 2l) using *S. typhimurium* TA98 and TA 100, respectively.

**Bacterial strains**
*S. typhimurium* TA98 and TA 100 bacterial strains were obtained from Food Biotechnology Laboratory, Life Sciences Institute, University of Agricultural Sciences and Veterinary Medicine Cluj-Napoca, Romania, cultured on Muller–Hinton Agar, stored at 4 °C and subcultured once a month. Sterile Erlenmeyer flasks containing 20 mL of Oxoid nutrient broth were inoculated with the tested bacteria strains and incubated at 37 °C at 150 rpm for 16 hours. The starting cell density of the bacterial cultures was 1–2×10⁹ bacteria/mL, with absorbances at 660 nm in the range of 1.2 to 1.4.

**Viability assay**
The viability assays were based on a non-statistical procedure for evaluating the spontaneous mutation [5] induced by saturated solutions of the new compounds 2c, 2i, and 2l, respectively, in dimethyl sulfoxide (DMSO) (2c, 79.8 mM; 2i, 163.6 mM; 2l, 37.5 mM).
The positive controls were the known mutagens 2-aminoanthracene or sodium azide, NaN₃. The negative control was DMSO.

**Mutagenicity test**
A mixture of 100 µL of the bacterial culture with or without 500 µL of S9 and 100 µL test compounds was poured onto the surface of minimal glucose agar plates. Viable cell colonies were scored after incubation at 37 °C for 72 h. The experiments were performed in duplicate. The mutagenic activity was assessed by the non-statistical procedure [6].

**Antimutagenicity test**
A mixture of 100 µL of the bacterial culture, 100 µL mutagen (2-aminoantracene, daunomycin for *S. typhimurium* TA98 or sodium azide for *S. typhimurium* TA100), with or without additional 500 µL of S9 and 100 µL test compound were poured onto minimal glucose plates and viable cells were scored after incubation at 37 °C for 72 h.
The inhibition of mutagenicity was calculated by using equation 1 [7]

\[
\text{% Inhibition} = [1 – \frac{N_1}{N_0}] \times 100
\]

where

\(N_1\) - number of revertants per plate in the presence of both, the mutagen and tested compound

\(N_0\) - number of revertants per plate in the positive control
The antimutagenic effect was considered weak or absent (inhibition up to 25%), moderate (25–40% inhibition) or strong (inhibition higher than 40%) [8]

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Figure S1. $^1$H NMR spectrum (600 MHz) for compound 2a in CDCl$_3$.

Figure S2. $^{13}$C NMR spectrum (150 MHz) for compound 2a in CDCl$_3$. 
Figure S3. 2D-NMR $^1$H/$^1$H COSY spectrum for compound 2a in CDCl$_3$. 
Figure S4. $^1$H NMR spectrum (600 MHz) for compound 2b in CDCl$_3$.

Figure S5. $^{13}$C NMR spectrum (150 MHz) for compound 2b in CDCl$_3$. 
Figure S6. 2D-NMR $^1$H/$^1$H COSY spectrum for compound 2b in CDCl$_3$.

Figure S7. $^1$H NMR spectrum (600 MHz) for compound 2c in CDCl$_3$. 
**Figure S8.** $^{13}$C NMR spectrum (150 MHz) for compound 2c in CDCl$_3$.

**Figure S9.** 2D-NMR $^1$H/$^1$H COSY spectrum for compound 2c in CDCl$_3$. 
Figure S10. $^1$H NMR spectrum (600 MHz) for compound 2e in CDCl$_3$.

Figure S11. $^{13}$C NMR spectrum (150 MHz) for compound 2e in CDCl$_3$. 

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Figure S12. 2D-NMR $^1$H/$^1$H COSY spectrum for compound 2e in CDCl$_3$.

Figure S13. $^1$H NMR spectrum (400 MHz) for compound 2g in CDCl$_3$. 
Figure S14. $^{13}$C NMR spectrum (125 MHz) for compound 2g in CDCl$_3$.

Figure S15. $^1$H NMR spectrum (400 MHz) for compound 2i in CDCl$_3$. 
Figure S16. $^{13}$C NMR spectrum (125 MHz) for compound 2i in CDCl$_3$.

Figure S17. 2D-NMR $^1$H/$^1$H COSY spectrum for compound 2i in CDCl$_3$. 
Figure S18. 2D-NMR $^1$H/$^{13}$C HMQC spectrum for compound 2i in CDCl$_3$.

Figure S19. $^1$H NMR spectrum (600 MHz) for compound 2j in CDCl$_3$. 
Figure S20. $^{13}$C NMR spectrum (150 MHz) for compound 2j in CDCl$_3$.

Figure S21. 2D-NMR $^1$H/$^1$H COSY spectrum for compound 2j in CDCl$_3$. 
Figure S22. $^1$H NMR spectrum (600 MHz) for compound 2k in DMSO-$d_6$.

Figure S23. $^{13}$C NMR spectrum (150 MHz) for compound 2k in DMSO-$d_6$. 
Figure S24. 2D-NMR $^1$H/$^1$H COSY spectrum for compound 2k in DMSO-$d_6$.

Figure S25. $^1$H NMR spectrum (400 MHz) for compound 2l in CDCl$_3$. 
Figure S26. $^{13}$C NMR spectrum (125 MHz) for compound 2I in CDCl$_3$. 