Aspartate\textsuperscript{112} is the Selectivity Filter of the Human Voltage Gated Proton Channel

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Abstract

The ion selectivity of pumps and channels is central to their ability to perform a multitude of functions. Here we investigate the mechanism of the extraordinary selectivity of the human voltage gated proton channel\textsuperscript{1}, hHV\textsubscript{1}. This selectivity is essential to its ability to regulate reactive oxygen species production by leukocytes\textsuperscript{2–4}, histamine secretion by basophils\textsuperscript{5}, sperm capacitation\textsuperscript{6}, and airway pH\textsuperscript{7}. The most selective ion channel known, H\textsubscript{V}1 shows no detectable permeability to other ions\textsuperscript{1}. Opposing classes of selectivity mechanisms postulate that (a) a titratable amino acid residue in the permeation pathway imparts proton selectivity\textsuperscript{1, 8–11}, or (b) water molecules “frozen” in a narrow pore conduct protons while excluding other ions\textsuperscript{12}. Here we identify Aspartate\textsuperscript{112} as a crucial component of the selectivity filter of hHV\textsubscript{1}. When a neutral amino acid replaced Asp\textsuperscript{112}, the mutant channel lost proton specificity and became anion selective or did not conduct. Only the glutamate mutant remained proton specific. Mutation of the nearby Asp\textsuperscript{185} did not impair proton selectivity, suggesting that Asp\textsuperscript{112} plays a unique role. Although histidine shuttles protons in other proteins, when histidine or lysine replaced Asp\textsuperscript{112}, the mutant channel was still anion permeable. Evidently, the proton specificity of hHV\textsubscript{1} requires an acidic group at the selectivity filter.

Voltage gated proton channels are considered specific (perfectly selective) for protons, because no evidence exists for permeation of anything but H\textsuperscript{+}. Specificity, combined with a
large deuterium isotope effect and extraordinarily strong temperature dependence of conduction suggests a permeation pathway more complex than a simple water wire, as exists in gramicidin. All proton conduction appears consistent with a hydrogen-bonded chain (HBC) mechanism; a HBC including a titratable group could explain several unique properties of Hv1, especially proton selectivity. Yet in a recent study, mutation of each titratable amino acid in all four transmembrane (TM) helices of Hv1 failed to abolish conduction. Thus, the mechanism producing proton selectivity remained unknown.

We noticed that a human gene, C15orf27 (of unknown function), contains a predicted voltage sensor domain (VSD) that shares 25% sequence identity and 52% similarity with the VSD of Hv1, and includes three Arg residues in the S4 TM helix that are conserved among all known Hv1 homologues. Phylogenetic analysis of VSD sequences reveals a group comprising Hv1, C15orf27, and voltage sensitive phosphatase (VSP) sequences separated early from the two phylogenetically distinct groups of depolarization activated VSDs described previously (Kv channels and Nav/Cav channels), supporting the modular evolution of VSD-containing proteins. Furthermore, Hv1 VSDs occupy a discrete lineage, distinct from those of VSP and C15orf27 orthologs.

When we cloned the C15orf27 gene and expressed the product in HEK-293 or COS-7 cells, the GFP-tagged protein localised at the plasma membrane, but we detected no currents beyond those in non-transfected cells. We reasoned that substitutions based on sequence elements that differ between Hv1 and C15orf27 should be structurally tolerated while revealing residues responsible for proton conduction. We therefore mutated residues that are perfectly conserved in 21 Hv1 family members and differ between C15orf27 and Hv1. We replaced five candidate residues in hv1 (D112, D185, N214, G215, and S219) with the corresponding residue in C15orf27. Four mutants exhibited large currents under whole-cell voltage clamp. The reversal potential, measured at several pHo and pHi, was close to the Nernst potential for protons, E_H, demonstrating proton selectivity. D112V mutants localised to the plasma membrane, but displayed no convincing current. Some D112V transfected HEK-293 or COS-7 cells (and non-transfected cells) exhibited small native proton currents. His140A/H193A double mutants, in which the two Zn binding His residues are neutralized, resemble WT, with similar ΔpH dependent gating, and V_rev near E_H. We expressed mutants in this Zn insensitive background (D112x/A/A) to distinguish their currents from native currents that are abolished by 100 μM Zn at pH 7.0. We tentatively concluded that Asp112 is crucial to proton conduction.

The absence of detectable currents in D112V led us to make other D112x substitutions. These mutants exhibited slowly activating outward currents upon depolarization that resembled hv1 currents. As reported previously, Asp112 mutation had little effect on the ΔpH dependence of gating. The proton conductance-voltage relationship of all D112x mutants shifted roughly -60 mV when pHo increased from 5.5 to 7.0, as in WT channels. Mutation of Asp112 did influence channel opening and closing kinetics.
Measurements of \( V_{\text{rev}} \) in Asp\(^{112} \) mutants revealed a marked departure from WT hH\( \text{V}_1 \) properties. At symmetrical pH 5.5, \( V_{\text{rev}} \) was near 0 mV (not shown). At pH\(_o\) 7.0, pH\(_i\) 5.5 (Fig. 2a, column 3), WT channels reversed near \( E_H \) (−87 mV), indicating proton selectivity. But for all mutants except D112E, \( V_{\text{rev}} \) was substantially positive to \( E_H \) (Fig. 2b), ranging from −58 mV (D112H) to −13 mV (D112N). Substitutions at Asp\(^{112} \) eliminated the proton specificity that distinguishes H\( \text{V}_1 \) from all other ion channels\(^1 \). A previous study described currents in D112A and D112N mutants\(^12 \) but did not report \( V_{\text{rev}} \).

We expected that loss of proton selectivity would result in nonselective permeation of cations. Surprisingly, \( V_{\text{rev}} \) did not change detectably when Na\(^+\), K\(^+\), N-methyl-D-glucamine\(^+\), or TEA\(^+\) replaced TMA\(^+\) (Table S4). To test anion vs. cation selectivity, we adopted the classical tactic of replacing a fraction of the bath solution with isotonic sucrose\(^19 \). The Nernst equation predicts that dilution of all extracellular ions except H\(^+\) and OH\(^−\) (leaving internal ion concentrations unchanged) will shift \( V_{\text{rev}} \) negatively for a cation selective channel, but positively for an anion selective one. Despite the 10-fold reduction of buffer concentration, direct measurement confirmed that pH remained constant.

Fig. 3 illustrates determination of \( V_{\text{rev}} \) from tail currents in a D112H transfected cell at pH 5.5/5.5 in CH\(_3\)SO\(_3\)\(^−\) (Fig. 3a) or Cl\(^−\) solution (Fig. 3c), and after 90% reduction of external ionic strength (Figs. 3b & 3d). Astonishingly, for all Asp\(^{112} \) mutants except D112K/A/A, sucrose shifted \( V_{\text{rev}} \) positively (Fig. 3f), indicating anion selectivity both in CH\(_3\)SO\(_3\)\(^−\) (Fig. 3e) and Cl\(^−\) solutions (Fig. 3f). For hH\( \text{V}_1 \) and D112E, \( V_{\text{rev}} \) did not change, reaffirming their proton specificity. Neutralization of a single Asp residue converts a proton channel into a predominantly anion selective channel. Thus, Asp\(^{112} \) mediates charge selectivity as well as proton selectivity.

To confirm anion permeability of Asp\(^{112} \) mutants, we replaced the main external anion, methanesulfonate\(^−\) (CH\(_3\)SO\(_3\)\(^−\)), with Cl\(^−\). Consistent with previous studies\(^1 \), \( V_{\text{rev}} \) in hH\( \text{V}_1 \) was unchanged. As shown in D112H (Fig. 3a vs. 3c), \( V_{\text{rev}} \) shifted negatively in Cl solutions in all mutants (except D112E), indicating that Cl\(^−\) is more permeant than the larger CH\(_3\)SO\(_3\)\(^−\) anion (Fig. 3g). That all conducting non-acidic mutants exhibited Cl\(^−\) permeability indicates that Asp\(^{112} \) mediates not only proton selectivity, but also charge selectivity. Currents were smaller than WT in cells expressing some mutants (Fig. S7), suggesting a smaller unitary conductance. Evidently, these channels conduct anions, but not very well.

Although the mutant channels have diminished selectivity, \( V_{\text{rev}} \) did shift negatively when pH\(_o\) increased from 5.5 to 7.0 (Fig. 2b). Because these solutions differ mainly in buffer species and concentrations of H\(^+\) and OH\(^−\), Asp\(^{112} \) mutants must have significant permeability to H\(^+\) and/or OH\(^−\). The Goldman-Hodgkin-Katz equation shows how \( V_{\text{rev}} \) depends on ion concentrations:

\[
V_{\text{rev}} = \frac{RT}{zF} \log \left( \frac{P_{\text{Cl}−}[\text{Cl}−]_{i} + P_{\text{CH}_3\text{SO}_3−}[\text{CH}_3\text{SO}_3−]_{i}}{P_{\text{Cl}−}[\text{Cl}−]_{o} + P_{\text{CH}_3\text{SO}_3−}[\text{CH}_3\text{SO}_3−]_{o}} \right) + P_{\text{OH}−}[\text{OH}−]_{i} + P_{H+}[H+]_{o}.
\]

\[\text{(1)}\]

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Ions with greater permeability dominate $V_{\text{rev}}$. Permeation of $H^+$ and $OH^-$ are difficult to distinguish because they have the same Nernst potential\(^1\). The data can be interpreted assuming permeation of either (Table S3), but the anion selectivity of Asp\(^{112}\) mutants and the pH dependence of sucrose effects (Figs. 3e & 3f) support $OH^-$ permeation. The relative permeability of conducting Asp\(^{112}\) mutants was $OH^-$ (or $H^+$) > $Cl^-$ > $CH_3SO_3^-$. Although Asp\(^{112}\) is essential to selectivity, other acidic groups might participate. We mutated Asp\(^{185}\), located in the presumed conduction pore (Fig. 1b)\(^{12,17}\). However, like D185M (Fig. 1b), D185V, D185A, and D185N remained proton selective (Fig. S8). Speaking against additive effects, the double mutant, D112N/D185M did not differ from D112N (Fig. S8).

Consistent with earlier predictions that a titratable amino acid provides the selectivity filter of H\(_{V1}\)\(^{1,8-11}\), only channels with acidic residues (Glu or Asp) at position 112 manifested proton specificity. Asp\(^{112}\) lies at the constriction of the presumed pore (Fig. 1c), a logical location for a selectivity filter, and just external to the postulated gating charge transfer centre\(^{20}\). Our original prediction envisioned selectivity arising from protonation/deprotonation of a residue during conduction, but other mechanisms are possible. For example, proton selectivity of the influenza A M\(_2\) viral proton channel has been explained by (a) immobilized water\(^{21}\), (b) successive proton transfer and release by His\(^{37}\) (refs.\(^{22,23}\)), or (c) delocalization of the proton among His\(^{37}\) and nearby waters\(^{24}\).

The Cl\(^-\) permeability of D112H was completely unexpected given strong precedents for His imparting proton selectivity to channels. Histidine shuttles protons in K\(^+\) or Na\(^+\) channel VSDs with Arg→His mutations\(^{25-27}\), in carbonic anhydrase\(^{28}\), and in M\(_2\) channels\(^{22,23}\). However, these molecules are not proton specific\(^{27,29}\). Evidently, His shuttles protons, but does not guarantee proton selectivity. In hH\(_{V1}\), Asp\(^{112}\) (or Glu\(^{112}\) in D112E) excludes anions, resulting in proton specific conduction. His may fail to exclude anions because it is cationic when protonated, whereas Glu and Asp are neutral.

The anion selectivity of neutral Asp\(^{112}\) mutants suggests that electrostatic forces due to the charge distribution in the rest of the channel deter cation permeation, and that the cation selectivity of the WT channel is due to the anionic charge of Asp\(^{112}\). Asp\(^{185}\) does not participate directly in selectivity (Fig. S8). VSP family members possess the equivalent of Asp\(^{112}\) (Fig. 1a), yet conduct no current\(^{30}\), illustrating that Asp\(^{112}\) requires a specific microenvironment to achieve selectivity. Although permeation of Cl\(^-\) and $CH_3SO_3^-$ suggests a wide pore in D112x mutants, local geometry might differ in WT channels due to the presence of anionic Asp\(^{112}\).

Regulation of voltage gating by ΔpH is distinct from permeation. Pathognomonic of H\(_{V1}\) is a strict correlation between the $g_{H-V}$ relationship and $V_{\text{rev}}$, in which $V_{\text{threshold}}$ shifts 40 mV/Unit change in ΔpH\(^8\). The ΔpH dependence persisted in mutants exhibiting shifted $g_{H-V}$ relationships\(^{12}\). Here we show uncoupling of $V_{\text{rev}}$ and voltage gating. Asp\(^{112}\) mutants retained normal ΔpH dependence (Fig. S5), despite the dissociation of $V_{\text{rev}}$ from ΔpH (Figs. 2 & S9). This uncoupling of pH control of gating from permeation speaks against any

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mechanism that invokes regulation by local proton concentration in the vicinity of S4 Arg residues. In summary, Asp is a critical component of the selectivity filter of hHV1, crucial to both proton selectivity and charge selectivity. That D112E was proton selective, but D112H conducted anions indicates that this proton channel requires an acid at the selectivity filter. That neutralization of nearby Asp did not affect selectivity suggests that Asp plays a unique role.

‘Methods Summary’

The pipette solution (also used externally) contained (mM) 130 TMACH3SO3, 2 MgCl2, 2 EGTA, 80 MES, titrated to pH 5.5 with ~20 TMAOH. In the pH 5.5 TMACl solution, TMACl replaced TMACH3SO3. Bath solutions at pH 7.0 had (mM) 90 TMACH3SO3 or TMACl, 3 CaCl2, 1 EGTA, 100 BES, and 36–40 TMAOH. For experiments with Zn2+, solutions contained PIPES without EGTA. Experiments were done at 20–25°C. Currents are shown without leak correction. Vrev data were corrected for liquid junctions potentials measured in each solution.

‘Methods’

Exhaustive searches to identify H1 homologs were performed using protein BLAST and PSI-BLAST. A sample of VSDs from K+, Na+, and Ca2+ channels (that open with depolarization like H1, and in addition one that opens with hyperpolarization), along with putative H1, VSP, and C15orf27 homologs were chosen. For cation channels, we sampled from the range of subfamilies, from the VSD repeats within Na+ and Ca2+ channels, and from the range of species. VSD sequences, including crystallized K+ channels (PDB ids 1ORS, 2R9R, and 2A79), were aligned using PromalS3D, which incorporates structural information, allowing high confidence identification of VSD boundaries. Sequences were trimmed to the VSD, realigned with PromalS3D, and the resulting alignment was analyzed with PhyML (maximum likelihood) and Protpars (maximum parsimony) at the Mobyle portal. Trees were visualized with TreeDyn and iTOL. Parsimony (not shown) and maximum likelihood trees had similar topology, including H1 and C15orf27 families separating into discrete branches. A homology model of the VSD of hHV1 was constructed as described previously.

The C15orf27 clone was PCR amplified from human cerebellum and subcloned into pcDNA3.1(+) expression vector (Invitrogen, Carlsbad, CA). The coding sequence of human H1 (HVCN1) was cloned into either pcDNA3.1(−) or pQBI25-fC3 (to make GFP-H1) vectors as described previously. Site directed mutants were created using the Stratagene Quickchange (Agilent, Santa Clara, CA) procedure according to the manufacturer’s instructions. All the positive clones were sequenced to confirm the presence of the introduced mutation. HEK-293 or, more often COS-7 cells were grown to ~80% confluency in 35 mm cultures dishes, usually by seeding cells 1 d ahead of transfection. Cells were transfected with 0.4–0.5 μg of the appropriate cDNA using Lipofectamine 2000 (Invitrogen). After 6 h at 37°C in 5% CO2, the cells were trypsinized and re-plated onto glass cover slips at low density for patch clamp recording the following day. We selected
green cells under fluorescence for recording. Patch clamp methods were described previously.18

The main pipette solution (also used externally) contained (in mM) 130 TMACH$_3$SO$_3$, 2 MgCl$_2$, 2 EGTA, 80 MES, titrated to pH 5.5 with ~20 TMAOH. In the pH 5.5 TMACl solution, TMACl replaced TMACH$_3$SO$_3$. Bath solutions at pH 7.0 had (mM) 90 TMACH$_3$SO$_3$ or TMACl, 3 CaCl$_2$, 1 EGTA, 100 BES, and 36–40 TMAOH. For experiments with Zn$^{2+}$, solutions contained PIPES buffer without EGTA. Experiments were done at 20–25°C. Currents are shown without leak correction. $V_{rev}$ data were corrected for liquid junctions potentials measured in each solution.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

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Figure 1. Identification of five key amino acids that differ in hHV1 and C15orf27, and the currents generated in a heterologous expression system by hHV1 mutants in which hHV1 residues were replaced by the corresponding amino acid in the non-conducting C15orf27.

a, Representative subset of multiple sequence alignment of 122 VSDs; only TM helices are shown. Gene families include: H_V1; voltage sensitive phosphatases; C15orf27; Ca\(^{2+}\) and Na\(^{+}\) channels; and K\(^{+}\) channels (cf. Fig. S1).

b, Location of the key amino acids in the open hHV1 channel VSD viewed from the side (membrane), based on a homology model\(^{17}\).

c, \(V_{rev}\) in the four conducting mutants is near \(E_H\) (dashed line), indicating proton selectivity. \(V_{rev}\) was measured using tail currents; in G215A \(V_{rev}\) was positive to threshold and was observed directly.

d, Voltage-clamp current families in cells expressing hHV1 mutants. Depolarizing pulses were applied in 10-mV increments from a holding voltage, \(V_{hold} = -40\) mV (D185M, D112V), −60 mV (G215A, S219P), or −90 mV (N214D), with the most positive pulse labelled. After membrane repolarisation, an inward “tail current” is seen as channels close (see inset for D185M); pH is given as pH\(_{o}/\text{pH}_{i}\). D112V exhibited no clear current.
Figure 2. Currents in Asp112 mutants resemble proton currents, but are not
a, Currents generated by WT, D112E, D112H, D112K/A/A, D112N/A/A, D112S, D112A/A/A, and D112F/A/A in COS-7 cells (pHi 5.5) at pHo 5.5 (column 1) or 7.0 (column 2), during families of pulses in 10 mV increments up to indicated voltages. Tail currents at pHo 7.0 (column 3) reveal that Vrev deviates from E_H, indicating loss of proton selectivity. At pHo 5.0 Vhold was −40 mV (−60 mV for WT). At pHo 7.0 Vhold was −40 mV (K, N, S), −50 mV (F), −60 mV (H, A), −80 mV (E), or −90 mV (WT); Vpre was −65 mV (WT), −40 mV (E), −10 mV (H), +50 mV (K), +40 mV (N, S), or +20 mV (A, F). Vrev (arrows) was determined from the amplitude and direction of tail current decay. For D112N, Vrev was above Vthreshold and was evident during pulse families. b, Shift in Vrev when the TMACH3SO3 bath solution was changed from pH 5.5 to 7.0. There is no difference between WT and D112E, but the shift in all other mutants is smaller than WT (p<0.001, by one-way ANOVA followed by Tukey’s test, n = 7, 4, 9, 8, 6, 7, 9, and 4. Dashed line shows E_H.
Figure 3. Dilution of ionic strength by 90% with isotonic sucrose shifted $V_{\text{rev}}$ positively, indicating that most Asp$^{112}$ mutants are anion selective

**a** Measurement of $V_{\text{rev}}$ by tail currents in a cell transfected with D112H at pH 5.5/5.5 and (b) after sucrose. **c**, $V_{\text{rev}}$ in the same cell in pH 5.5 Cl$^-$ solution, and (d) after sucrose. Arrows indicate zero current. $V_{\text{hold}} = -40$ mV, $V_{\text{pre}} = +60$ mV. **e**, Mean shifts of $V_{\text{rev}}$ with decreasing ionic strength in CH$_3$SO$_3^-$ solutions or (f) in Cl$^-$ solutions. Each value was determined in 3–6 cells. X = not done. **g**, Shifts of $V_{\text{rev}}$ when CH$_3$SO$_3^-$ was replaced by Cl$^-$. Values for WT and D112E do not differ significantly from 0 mV. For all anion selective mutants except D112N, the difference between shifts at pH 5.5 and 7.0 was significant ($p<0.001$, one-way ANOVA followed by Tukey’s test; $n = 3–8$). Error bars in **e–g** are s.e.