Evaluation of Oat Extracts on the Efficiency of Lime in Soil

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ABSTRACT

The low mobility of the soluble components of surface-applied lime, limits their ability to reduce subsoil acidity in variable charge soils. Laboratory experiments were conducted with brazilian Oxisol to evaluate the effect of oat extracts on the mobility of surface applied calcium in form of CaCO₃. The following oat cultivars were evaluated: Argentina 5VL3, UPF 90H400-2, IAPAR 61, IA96101-b, SI 83400, EMBRAPA 29, LD 9102, FAPA 1, ER 93247, ER 90148, ER 89144, ER 93152, Alpha 94124, Alpha 94206 and Preta comum. Oat extracts and lime were applied at rates of 10 and 5 Mg ha⁻¹, respectively. Effect of lime without plant extracts on soil acidity was limited in the upper 10 cm of the profile. Lime in the presence of oat extracts increased pH and Caₓ and decreased Alₓ about 20 cm deep. The SI83400 and UPF90H400-2 were the best oat extracts as Ca-carrier into the acid soil profile. The results suggested that soluble organic compounds from oat extracts are mobile and capable of detoxifying acid subsoil, by increasing liming efficiency.

Key words: subsoil acidity, Ca-leaching, plant residue.

INTRODUCTION

Large areas in Brazil contain acid soils and some of which have subsoil acidity (Olmos & Camargo, 1976). Acid subsoils contain toxic levels of Al or deficient amount of Ca which restrict root proliferation (Ritchey, 1981). The major consequence of subsoil acidity is drought stress. The efficacy of surface-applied lime in neutralizing subsoil acidity is uncertain. In some case no effect was observed (Pavan et al., 1984) in others lime effect is quite rapid (Oliveira & Pavan, 1996).

Many examples have been presented in the literature to demonstrate that alkalinity can be leached to subsoil by the additions of organic compounds on the soil surface. Wright et al. (1985) showed that dolomitic lime in the presence of cow manure increased subsoil pH and decreased exchangeable Al whereas lime alone had no effect. Oliveira & Pavan (1996) demonstrated that lime applied on soil surface under no tillage with high amount of plant residue serving as mulch, had its effects extended to a depth of 60cm, increasing pH and Ca and decreasing Al contents. Miyazawa et al. (1998), Pavan & Miyazawa (1998) and Ziglio et al. (1999) showed that black oat residue improved the efficiency of liming, increasing pH and Ca and reducing Al in the subsoil profile. They also reported that wheat residue had no effect in lime efficiency. They concluded that a range of low molecular weight organic acids are produced in soil from the decomposition of organic matter. Although these organic acids are considered to be short-lived in soils, their continual production makes their anions important in acid soil chemistry.

The objective of this work was to evaluate the efficiency of oat extracts on the effect of soil surface-applied lime.

MATERIAL AND METHODS

Plant sampling

Fifteen black and white oat cultivars were planted in the Instituto Agronômico do Paraná (IAPAR) experimental station at Londrina, Brazil (23° 20’S and 51° 08’W) during 1998 spring season. The cultivars were: Argentina 5VL3, UPF 90H400-2,
IAPAR 61, IA96101-b, SI 83400, EMBRAPA 29, LD 9102, FAPA 1, ER 93247, ER 90148, ER 89144, ER 93152, Alpha 94124, Alpha 94206 and Preta comum. At the maximum growing stage, shoot samples were collected from each cultivar, dried at 65°C during 72h in a forced drauth oven, finely ground to pass 1 mm sieve and stored in paper bags. Sub samples were then chemically analyzed by standard procedure used at IAPAR (Miyazawa et al., 1992). Plant extract solution was obtained by the following procedure: transfer 3.0 g (equivalent to 10 Mg ha⁻¹) of dried plant material to centrifuge tube, add 150 ml of water (corresponding to 1 soil porous volume-pv), shake for 8 hours and filter.

**Soil sampling**

A bulk of soil material was collected from the 0-20 cm horizon of an uncultivated red dark latosol (Oxisol). The soil material was air dried and ground to pass through a 2mm screen before being treated. Sub sample was analyzed by the standard procedure used at IAPAR (Pavan et al., 1992). It had an original pHCaCl₂ value of 4.10; exchangeable Al, Ca, Mg, and K contents of 11.1, 3.70, 1.70 and 0.50 mmol dm⁻³, respectively; total acidity (H⁺+Al) of 62.0 mmol dm⁻³; and total carbon content of 7.64 g kg⁻¹. The clay content was 27%, silt 2% and sand 71%.

**Experimental treatments**

Soil sample was transferred to a PVC column (25 cm high and 4 cm of diameter) and compacted to a homogeneous bulk density (mean 1g cm⁻³) corresponding to natural conditions. Calcium carbonate was added on the soil surface in an amount required to neutralize 200% of the total soil acidity (H⁺+Al) at a depth of 5 cm. Plant extract solutions were added at a rate of 1 ml min⁻¹. Then 150ml of deionized water was added (1pv). This was done three times, to a total of 3pv. Each time, the leachate (1pv) was collected in plastic bottles attached at the bottom of the soil column. All treatments had three replicates in a completely randomized block design. The leachates were analyzed for pH, Al and Ca. Soil samples were taken from each replicate at the following depths: 0-5, 5-10, 10-15, 15-20, and 20-25 cm. The soil samples were air dried, ground to pass 2mm sieve, and analyzed for pH and exchangeable Al and Ca. The soil pH was determined in 0.01 mol L⁻¹ CaCl₂ suspension (1:2.5 soil:solution ratio) after shaking for 1h; exchangeable Al and Ca were extracted with N KCl solution in a 1:10 soil:solution ratio and 10 minutes shaking time. Al was determined by titration with standardized 0.015N NaOH solution using bromotymol blue indicator and Ca by atomic absorption spectroscopy.

**RESULTS AND DISCUSSION**

The soil pH, Ca, and Al by depth for the control treatment are shown in figures 1, 2, and 3, respectively. These figures show the effects of control and the most and the least efficient oat extracts. The other extracts presented intermediate effects. Unamended soil profile (no lime, no extract) had lower pH (average 4.1), lower Ca (average 3.5 mmol dm⁻³) and higher Al (average 6.7 mmol dm⁻³). These chemical characteristics are believed to restrict root proliferation and crops are likely to respond to lime.

Additions of lime alone increased pH and Ca, and decreased Al in the up 10cm surface of the profile only (data not presented). These results confirm the low efficacy of surface applied lime in reducing subsoil acidity in variable charge soil, corroborating data presented in previous study with brazilian soils (Pavan et al., 1984). Application of lime in the presence of oat extracts markedly increased soil pH and Ca down to 20cm depth and neutralized Al down to 15cm depth. These results indeed support the view that soluble organic compounds in soil are mobile and capable of detoxifying acid subsoils. Miyazawa et al. (1992) demonstrated that organic acids can detoxify Al³⁺ to an acid sensitive wheat cultivar. The SI83400 and UPF 90H400-2 were the best oat extracts as Ca-carrier deeper in the soil profile, and ER 89144 was the least efficient. The UPF 90H400-2 was also the most efficient oat extract in reducing Al deeper in the soil profile (Figure 3).

![Figure 1 - The effects of oat extracts on soil pH.](image-url)
The results concerning the chemical analysis of the leachate (pH, Ca and Al) are presented in figures 4, 5, and 6. Unamended soil profile and limed soil surface without plant extracts showed low concentration of Ca in the leachate. Oat extracts were separated in two groups: the least (A) and the most efficient (B) in increasing pH, Ca and Al, and they were compared with control. Increasing pH increased pH (dilution factor) and decreased leachate Ca and Al. The group A for pH were: ALPHA 94206, ALPHA 94124, Preta comum, ER 93152, ER 89144, LD 9102, EMBRAPA 29, FAPA 1, ER 93247 and ER 90148. The group B for pH were: IAPAR 61, IA 96101-b, UPF 90H400-2, Argentina 5VL3 and SI 83400. The group A for Ca were: IA 96101-b, SI 83400, UPF 90H400-2, IAPAR 61 and Argentina 5 VL3. The group B for Ca were: FAPA 1, ER 90148, ER 93427, Preta comum, ALPHA 94214, LD 9102, ALPHA 94206, EMBRAPA 29, ER 93152 and ER 89144. The group A for Al were: IA 96101-b, IAPAR 61, UPF 90H400-2, SI 83400, Argentina 5VL3, ALPHA 94206, Preta comum, ALPHA 94124, ER 93152 and FAPA 1.

The group B for Al were: ER 90148, ER 89144, LD 9102, EMBRAPA 29 and ER 93247. Al was not detected in the leachate, for unamended soil profile. These results show again that surface lime application on variable charge soil presents minimal vertical movement. However liming soil surface with plant extracts enhanced leaching losses of Ca and Al. The capacity of oat extract as Ca-carrier followed the order: SI83400 = UPF 90H400-2 > all the others. The capacity of oat extracts to detoxify Al followed the order: UPF 90H400-2 > SI83400 > IA 96101-b > IAPAR 61 > Argentina 5VL3 > LD 9102 > ER 93247 = FAPA 1 > EMBRAPA 29 = ALPHA 94124 = ER 93125 = ER 90148 > ALPHA 94206 > ER 89144 > Preta comum.
CONCLUSIONS

Soluble organic plant compounds from oat extracts improve the efficiency of surface-applied lime. The SIB3400 and UPF 90H400-2 oat extracts were the most efficient as Ca-carrier in soil.

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