Fast photodynamics of azobenzene probed by scanning excited-state potential energy surfaces using slow spectroscopy

Tan, E.M.M.; Amirjalayer, S.; Smolarek, S.; Vdovin, A.; Zerbetto, F.; Buma, W.J.

DOI
10.1038/ncomms6860

Publication date
2015

Document Version
Final published version

Published in
Nature Communications

License
CC BY

Citation for published version (APA):
Tan, E. M. M., Amirjalayer, S., Smolarek, S., Vdovin, A., Zerbetto, F., & Buma, W. J. (2015). Fast photodynamics of azobenzene probed by scanning excited-state potential energy surfaces using slow spectroscopy. Nature Communications, 6, [5860]. https://doi.org/10.1038/ncomms6860

General rights
It is not permitted to download or to forward/distribute the text or part of it without the consent of the author(s) and/or copyright holder(s), other than for strictly personal, individual use, unless the work is under an open content license (like Creative Commons).

Disclaimer/Complaints regulations
If you believe that digital publication of certain material infringes any of your rights or (privacy) interests, please let the Library know, stating your reasons. In case of a legitimate complaint, the Library will make the material inaccessible and/or remove it from the website. Please Ask the Library: https://uba.uva.nl/en/contact, or a letter to: Library of the University of Amsterdam, Secretariat, Singel 425, 1012 WP Amsterdam, The Netherlands. You will be contacted as soon as possible.

UvA-DARE is a service provided by the library of the University of Amsterdam (https://dare.uva.nl)
Fast photodynamics of azobenzene probed by scanning excited-state potential energy surfaces using slow spectroscopy

Eric M.M. Tan1, Saeed Amirjalayer1,†, Szymon Smolarek1,†, Alexander Vdovin1,†, Francesco Zerbetto2 & Wybren Jan Buma1

Azobenzene, a versatile and polymorphic molecule, has been extensively and successfully used for photoswitching applications. The debate over its photoisomerization mechanism leveraged on the computational scrutiny with ever-increasing levels of theory. However, the most resolved absorption spectrum for the transition to \( S_1(\pi\pi^*) \) has not followed the computational advances and is more than half a century old. Here, using jet-cooled molecular beam and multiphoton ionization techniques we report the first high-resolution spectra of \( S_1(\pi\pi^*) \) and \( S_2(\pi\pi^*) \). The photophysical characterization reveals directly the structural changes upon excitation and the timescales of dynamical processes. For \( S_1(\pi\pi^*) \), we find that changes in the hybridization of the nitrogen atoms are the driving force that triggers isomerization. In combination with quantum chemical calculations we conclude that photoisomerization occurs along an inversion-assisted torsional pathway with a barrier of \( \sim 2 \text{ kcal mol}^{-1} \). This methodology can be extended to photoresponsive molecular systems so far deemed non-accessible to high-resolution spectroscopy.
ight offers unique opportunities to address molecular systems non-invasively, in a highly selective manner, and with high space and time resolution. Compounds whose shape, biological and chemical activity, dielectric constant and refractive index can be switched reversibly under the control of light thus have found extensive applications in areas that range from molecular probes to molecular nanotechnology and photopharmacology\(^1\)–\(^4\). Azobenzene and its derivatives are one of the primary examples of such compounds. Their favourable properties such as photostability and functional activity under a wide range of experimental conditions have in many aspects paved the way for the further development and application of photochromic systems.

At the basis of the functional behaviour of azobenzene is the reversible (trans) (E)→(cis) (Z) isomerization of the N=N bond upon photoexcitation. The absorption spectrum of azobenzene has therefore long attracted attention. Its low-energy part displays two bands, a weak band in the visible region (\(\sim 430 \text{ nm} \)) associated with excitation of the formally dipole-forbidden \(\text{B}_u (n^*\pi^*)\) state and a strong band in the ultraviolet region (\(\sim 320 \text{ nm} \)) associated with absorption of the dipole-allowed \(\text{B}_g (\pi^*\pi^*)\) state\(^5\). One of the intriguing aspects of azobenzene photoisomerization is the dependence of the quantum yield on the excitation wavelength\(^6\)–\(^7\). In principle, isomerization in azobenzene can take place along two different pathways, one that involves torsional motion around the N=N double bond, and another in which in-plane inversion of one of the nitrogen centres takes place (Fig. 1). To accommodate the observed violation of Kasha’s rule, it was therefore originally proposed that isomerization proceeds along two different pathways in the \(\pi^*\) and \(\pi\pi^*\) states\(^5,8,9\). Over the years, extensive experimental\(^10\)–\(^19\) and theoretical\(^20\)–\(^29\) efforts have been made to elucidate the contribution of these pathways to the photoisomerization dynamics. As yet, however, there is still no consensus on the pathway, and also the interpretation of many other aspects of the excited-state dynamics are still a matter of extensive debate.

At the basis of a fundamental understanding of the photodynamics of azobenzene is a detailed knowledge on the potential energy surfaces of the relevant excited states, preferably under isolated molecule conditions. Such studies provide detailed insight on the forces that act on the molecular structure after excitation from the ground state, map out which normal coordinates are involved in the structural dynamics, reveal whether energy barriers are involved and provide a direct benchmark for theoretical calculations\(^30\)–\(^33\). Moreover, they must necessarily be the starting point for a further appreciation of the experimentally observed dependence of the photodynamics on the environment and substituents. As yet, however, azobenzene has predominantly been studied in the time domain with ultrafast laser techniques, at room temperature, and under non-isolated conditions. Under such conditions, temporal resolution is achieved at the expense of spectral resolution. As a result, the vibrational level structure of the excited state—which provides a direct fingerprint of the potential energy surface along the various normal coordinates—is not accessible.

Here, we show that advanced high-resolution two- and three-colour multiphoton ionization laser spectroscopic techniques using nanosecond lasers and molecular beams can be successfully applied to scan the potential energy surfaces of the \(S_1(n\pi^*)\) and \(S_2(\pi\pi^*)\) states of trans-azobenzene and measure the vibrational structure. Such an approach is well established, but its application to azobenzene appears to contradict a priori the conditions under which such studies are normally performed. Excitation of the \(S_2(\pi\pi^*)\) is formally dipole-forbidden, and only becomes allowed when the state is vibronically coupled to another dipole-allowed excited state. In combination with an excited-state lifetime that has been reported to be in the low-psicocone range\(^9\), resonance enhancement of the ionization yield is thus expected to be very hard to detect. Excitation of the \(S_2(\pi\pi^*)\) state is fully allowed, but for this state the reported lifetime of \(\sim 170 \text{ fs} \) in the gas phase\(^9\) severely impedes observation of resonance enhancement, and, if observed at all, lifetime broadening of resonances would be taken to preclude the detection of vibrationally resolved spectra. Despite these unfavourable conditions, we find that high-resolution spectra of both states can be obtained, and that these spectra provide a direct view on the photophysics and photochemistry of trans-azobenzene.

**Results**

**Spectroscopy and dynamics of the \(S_1(n\pi^*)\) state.** Figure 2a shows the \(S_1(n\pi^*)\)←\(S_0\) excitation spectrum of trans-azobenzene in the 18,250–20,750 cm\(^{-1}\) region, as detected by \((1+1')\) resonance-enhanced two-photon ionization (RE2PI) spectroscopy. Compared with the best-resolved absorption spectrum reported so far, which was obtained in mixed crystals\(^34\), the resolution of the present spectrum is several orders of magnitude better. It thereby allows for the first time a detailed analysis of the changes in molecular structure upon excitation and their implications on the photochemical activity of the molecule. Experiments and quantum chemical calculations indicate that the molecule retains \(C_{2h}\) symmetry in both the ground and \(S_1(n\pi^*)\) state. This implies that the \(S_1(n\pi^*)\) \((1^1\text{B}_g)\)←\(S_0\) \((1^1\text{A}_g)\) transition is formally dipole-forbidden. The spectrum observed in Fig. 2a must therefore be built upon false origins involving vibrational modes of either \(a_0\) or \(b_0\) symmetry\(^34\) that couple the \(1^1\text{B}_g\) state to higher-lying excited singlet states of \(\text{B}_0\) or \(\text{A}_0\) symmetry. The first electronically excited states of \(\text{B}_0\) and \(\text{A}_0\) symmetry are \(S_2\) and \(S_0\) respectively, with calculated oscillator strengths of 0.616 and 0.008 (ref. 24). On the basis of this large difference in oscillator strengths, it is reasonable to assume that the false origins in the \(S_1(n\pi^*)\)←\(S_0\) excitation spectrum involve \(a_0\) vibrations, as predicted previously\(^35\) and confirmed by our experimental data (vide infra).

Analysis of the excitation spectrum leads to the conclusion that it consists of several false origins upon which vibrational progressions involving totally symmetric \(a_0\) modes are built (see Supplementary Table 1), in particular \(v_{23}\) (188 cm\(^{-1}\)), \(v_{22}\) (294 cm\(^{-1}\)), \(v_{20}\) (653 cm\(^{-1}\)) and \(v_{19}\) (853 cm\(^{-1}\)). Importantly, we observe that \(v_{23}\), which contains a significant amount of the
NNC bend character, is the most active totally symmetric mode. This mode is directly associated with in-plane inversion of the two nitrogen atoms. Its motion thus relates directly to one of the suggested photoisomerization mechanisms, albeit that the actual photoisomerization process would involve a non-totally symmetric inversion mode. Nevertheless, the dominant activity of $v_{33}$ is significant, as it indicates that the hybridization and the related angles of the nitrogen atoms change considerably upon excitation. The importance of geometry changes along the inversion coordinate is also reflected in the activity of $v_{229}$ the second-most active mode. This mode involves predominantly CCC bending motions of the phenyl rings, but for a large part also NNC-bending motions.

Figure 2b displays the $S_1(n\pi^*) \leftrightarrow S_0$ excitation spectrum predicted within the Herzberg–Teller approximation. If we consider for the moment only the vibrational progressions involving totally symmetric $a_g$ modes, qualitatively good agreement is observed between experimental and predicted intensities (see also Supplementary Table 4), although quantitatively the differences are larger than what one would have expected on the basis of studies of similar complexity. In general, such differences indicate that the geometry changes upon excitation are not predicted accurately enough. To get an idea of the geometry changes that would lead to consistency with the experimentally observed excitation spectrum, we have employed the observed intensities of $v_{ij}(a_g^1)$ transitions to ‘reconstruct’ the equilibrium geometry of the $S_1(n\pi^*)$ state. In our experiments, we primarily observe Franck–Condon activity of bending vibrations and not of CC and CN stretch vibrations, which, as will be argued below, is attributed to the opening up of these vibrational energies. As a result, the reconstructed geometry primarily reflects changes in the bond angles and not in bond lengths. Table 1 reports the most relevant geometrical parameters of ground and excited states as originally predicted by our calculations—which are in excellent agreement with previous calculations—and as reconstructed from the excitation spectrum. This table indicates that the change in the NNC bond angle is considerably smaller than what is predicted by the calculations.

Above we have argued that the false origins are in principle associated with $a_u$ or $b_u$ vibrations. Consideration of the frequency intervals between the false origins and comparison with calculated vibrational frequencies rapidly leads to two conclusions. First, we find that these frequency intervals are in excellent agreement with what one would expect for the manifold of $a_u$ vibrations, but cannot be reconciled with the frequencies of $b_u$ vibrations (see Supplementary Table 3). Further support for an assignment to $a_u$ vibrations comes from simulations of the rotational contours of the bands (vide infra). These contours can be reproduced very well under the assumption of a $B_g (B_g \otimes a_u) \rightarrow A_g$ vibronic transition, but completely fail for an $A_g (B_g \otimes b_u) \rightarrow A_g$ vibronic transition. We conclude that the $S_1(n\pi^*) \leftrightarrow S_0$ transition obtains its intensity by vibronic coupling with the $1^aB_u$ state, in agreement with a priori expectations. Second, it has to be concluded that the strong band at $18,493.8 \text{ cm}^{-1}$ is the lowest-energy false origin and is associated with $v_{44}$ the lowest-frequency $a_u$ vibration. The true $S_1(n\pi^*) \leftrightarrow S_0$ 0–0 origin transition is not observed, but we can obtain a reasonably accurate estimate of its value by two different approaches, which both lead to the conclusion that it occurs at $18,471.7 \text{ cm}^{-1}$ (see Supplementary Notes 1 and 2).

The $S_1(n\pi^*) \leftrightarrow S_0$ excitation spectrum does not show only the $44^2$ false origin but also false origins associated with the $44^1$ and $44^3$ transitions. These transitions are forbidden within the harmonic oscillator approximation, and can only obtain intensity by anharmonicity or Duschinsky mixing, that is, normal-mode rotation upon excitation. As the energy differences between the bands indicate that the anharmonicity is rather small, one has to conclude that Duschinsky mixing is the principal reason for the appearance of the overtone false origins. Quantum chemical calculations confirm this conclusion: they show that $v_{44}$ and $v_{43}$ become extensively mixed upon excitation.

Apart from $v_{44}$, several other $a_u$ modes serve as false origins as well, most notably $v_{43}$, $v_{42}$, $v_{41}$, $v_{40}$ and $v_{47}$ whose transitions are displaced from the $S_1(n\pi^*) \leftrightarrow S_0$ false origin by $29.9$, $210.3$, $388.3$, $459.0$ and $786.4 \text{ cm}^{-1}$, respectively. Comparison of the relative intensities of the false origins (see Supplementary Table 5) shows that $v_{44}$ is by far the most effective in inducing intensity in the $S_1(n\pi^*) \leftrightarrow S_0$ transition. It is well known that vibronic coupling leads to changes in the frequency of the coupling mode. Compared with the uncoupled case, the frequency of the coupling mode is reduced in the lower state and increased in the upper state. Theory thus makes one expect that the relative frequency difference in the two states is a good measure for the vibronic coupling strength. Indeed, we find that the frequency of $v_{44}$ in $S_1$ and $S_2$ differs by more than a factor of two, while the other modes show much more moderate differences (see Supplementary Table 3). The conclusion that the $S_1(n\pi^*) \leftrightarrow S_0$ transition is to a dominant extent enabled by $v_{44}$ is an important one. This mode is essentially a pure CNNC torsional mode, the mode along which isomerization has been proposed to occur. The overall picture that thus emerges from the present isolated molecule spectrum is one in which the $S_1(n\pi^*)$ state can only be accessed if simultaneously an $a_u$ vibrational mode is excited, and that the most effective way to do so is to impart energy into the mode that involves torsion around the N=N double bond. It should be kept in mind that the present experiments have been performed under conditions in which vibrational levels in the ground state are not populated. At room temperature, however, low-frequency vibrations will be excited as well. Under such
Figure 3 | Rotational contour of bands in the $S_1(n\pi^*)$→$S_0$ excitation spectrum near the forbidden 0–0 transition. (a) Experimentally observed contours. (b) Simulated contours assuming a vibronic $B_u\rightarrow A_g$ transition, a rotational temperature of 9 K and a homogeneous linewidth for individual rotational transitions of $0.4\,\text{cm}^{-1}$.

Figure 4 | Influence of excitation energy on linewidths. Dependence of linewidths of bands in the $S_1(n\pi^*)$→$S_0$ excitation spectrum on excitation energy, the drawn line serving as a guide to the eye. For low excitation energies, bands have rather similar linewidths, but these start to increase markedly above $\sim 19,250$–$19,500\,\text{cm}^{-1}$. This increase gives evidence for an additional non-radiative decay channel that opens up for excess energies of $\sim 2\,\text{kcal mol}^{-1}$.

Table 1 | Relevant geometrical parameters of trans-azobenzene in the $S_0$, $S_1(n\pi^*)$ and $S_2(\pi\pi^*)$ states as determined from the experiment and predicted by quantum chemical calculations.

|                  | $S_0$ experiment | $S_1$ theory† | $S_1$ reconstructed‡,‡ | $S_2$ theory‡ | $S_2$ reconstructed‡ |
|------------------|------------------|---------------|-------------------------|---------------|----------------------|
| $d(C-N)$ (Å)     | 1.428            | 1.358         | 1.365                   | 1.369         | 1.373                |
| $d(N=N)$ (Å)     | 1.260            | 1.251         | 1.262                   | 1.351         | 1.324                |
| $\angle N=N-C$ (°) | 113.7           | 130.5         | 124.3                   | 111.5         | 112.2                |
| $\angle C-C-N$ (°) | 124.8           | 122.2         | 123.3                   | 124.4         | 124.6                |

*Structural parameters taken from gas-phase electron diffraction.†Present work.‡Determined from experimental intensities of $v_1(a_u)$ transitions for modes $i=23$–$19$, and the intensities predicted by the calculations for modes $i=18$–$6$ (see Supplementary Table 4).

The presence of a decay channel that only comes into play above a certain threshold energy explains the apparent mismatch between the lifetime measured in the present experiments and in solution. Time-resolved transient absorption and fluorescence experiments in solution differ in two aspects from the present experiments. First and most importantly, in solution molecules are excited at or near the absorption maximum, leading to excitation energies that are at least some 2,000 cm$^{-1}$ above the 0–0 transition determined here. Second, the present experiments have been performed at an effective rotational temperature of 9 K—the vibrational temperature admittedly being somewhat higher—while solution-phase experiments have been performed at room temperature. Both conditions lead to a situation in which the solution-phase experiments’ vibronic levels are excited in a
region well above the barrier for accessing the non-radiative decay channel discussed above. On a timescale of a few ps the effects of vibrational cooling are still limited\(^1\), and one would therefore expect that in the solution-phase experiments decay times are observed that are near the values estimated here for excitation of high-lying vibrionic levels. This expectation is borne out by our experiments: the reported range of decay times of 0.6–2.6 ps corresponds to linewidths of 2–9 cm\(^{-1}\).

The photoisomerization mechanism of azobenzene after excitation of the S\(_1(\pi\pi^*)\) state is still a subject of extensive debate. In the present study, we do not have direct access to the reactive part of the potential energy surface, but we do notice that our observations agree remarkably well with the relevant features of the potential energy surface as emerging from high-level \textit{ab initio} calculations\(^{24,26}\). This pertains in particular to the presence of a \(\sim 750–1,000\) cm\(^{-1}\) \((2.1–2.9\) kcal mol\(^{-1}\)) barrier for accessing an additional non-radiative decay channel. Calculations find on the potential energy surface of the S\(_1(\pi\pi^*)\) state a transition state with an activation energy of about 2 kcal mol\(^{-1}\).

Conical intersections of S\(_0\) and S\(_1\) along the torsional pathway are found at more or less the same energy, while conical intersections along the inversion pathway occur at much higher energies\(^{2,24}\). We thus conclude that the experimentally observed barrier is in excellent agreement with the predicted activation energy, while the experimental conclusion that above the barrier the excited state decays ultrafast to the electronic ground state is entirely consistent with an S\(_0\)/S\(_1\) conical intersection at similar energies as the activation energy. Taking into account that both experiment and theory agree that excitation is accompanied by a significant change of the NNC bond angle, it would appear that for the isolated molecule photoisomerization predominantly takes place via an inversion-assisted torsional pathway\(^3\) in which large changes in the CNNC dihedral angle occur simultaneously with smaller, but significant changes in the NNC angles. Importantly, changes along the CNNC torsional coordinate are intrinsically already prompted by the fact that excitation of the S\(_1(\pi\pi^*)\) state occurs predominantly by simultaneously exciting the N = N double-bond torsional vibration.

**Spectroscopy and dynamics of the S\(_2(\pi\pi^*)\) state.** The excitation spectrum of jet-cooled \textit{trans}-azobenzene to the second excited state is shown in Fig. 5a. We find that the S\(_2(\pi\pi^*)\) – S\(_0\) transition is located at 30,096.6 ± 0.2 cm\(^{-1}\). Qualitatively, the spectrum is similar to previously reported spectra of azobenzene in dibenzyl crystal\(^{44}\) and in the vapour phase\(^{45,46}\), but the resolution is significantly higher. This resolution in fact enables us to conclude that the widths of the bands are lifetime limited. Fits of the origin band with a Lorentzian find a width of 32.0 ± 1.0 cm\(^{-1}\) corresponding to a lifetime of 166 ± 5 fs, which is in excellent agreement with the lifetime reported in femtosecond time-resolved photoelectron spectroscopic studies (170 fs)\(^{9}\), and time-resolved fluorescence studies (110 fs)\(^{47}\).

Figure 5a shows that the most active mode in the S\(_2(\pi\pi^*)\) – S\(_0\) spectrum is the \(v_{\text{C}}\) (223 cm\(^{-1}\)), similar to what is observed for excitation of the S\(_1(\pi\pi^*)\) state. In agreement with the previous low-resolution studies\(^{44}\), we also see a prominent role of the \(v_{\text{C}}\) (1,315 cm\(^{-1}\)). However, at the same time the present resolution enables us to observe the activity of many other modes that so far have not been reported, and thereby to determine in greater detail the structural changes occurring upon excitation to this state. Detailed assignment of the bands in the excitation spectrum (see Supplementary Table 2) shows that they nearly all involve fundamentals of totally symmetric vibrations and their combinations. The only exception is the low-intensity band at 103 cm\(^{-1}\) from the origin, which also appears in combination with other bands. This band must be attributed to the 44\(_2\) transition. The activity of this mode indicates that it changes considerably upon excitation, as was also concluded for the S\(_1(\pi\pi^*)\) state.

Comparison of the experimental spectrum with the theoretically predicted spectrum (Fig. 5b) shows very good agreement, and indicates that the broadening of the experimental spectrum at higher energy is due to spectral congestion. In contrast to simulations of the S\(_1(\pi\pi^*)\) – S\(_0\) spectrum, the predicted intensities now follow quite closely the experimentally observed intensities. Accordingly, we find that reconstruction of the equilibrium geometry of the S\(_2(\pi\pi^*)\) state from the experimentally observed intensities leads to minor changes in bond lengths and bond angles (Table 1).

The relaxation pathways and dynamics of the S\(_2(\pi\pi^*)\) state have been the subject of a large amount of experimental and theoretical studies. In several of these studies, it has been suggested that other electronically excited states are involved in the decay of the S\(_2(\pi\pi^*)\) state\(^{24,28}\). Generally, different electronic states have different ionization cross-sections. One might thus see signatures of an excitation-energy-dependent involvement of other electronically excited states by comparing the (1 + 1) RE2PI excitation spectrum—for which intensities depend on both excitation and ionization cross-section—with the absorption spectrum of the S\(_2(\pi\pi^*)\) state obtained by ultraviolet–ultraviolet depletion spectroscopy for which intensities only depend on the
excitation cross-section. However, comparison of the ultraviolet–ultraviolet depletion spectrum shown in Fig. 5c with the RE2PI spectrum depicted in Fig. 5a does not give any indication of significant differences between the two spectra. This similarity also precludes significant contributions from decay channels leading to states that cannot be ionized, as would occur upon internal conversion to the ground state. We therefore conclude that our observations are consistent with an efficient ultrafast S1 ← S2 internal conversion, but cannot make definite statements on further details of this pathway.

Discussion

High-resolution nanosecond laser spectroscopy has enabled us to scan the potential energy surfaces of electronically excited states of trans-azobenzene. These studies have unveiled key aspects of how the structure of the molecule is changed upon photoexcitation, and how photoisomerization takes place. From these studies a picture has been established in which (a) excitation of the S1(nπ*) state is primarily accompanied by in-plane changes in the hybridization angles around the nitrogen atoms, (b) excitation of this dipole-forbidden state occurs most effectively if it involves the CNNC torsional mode, the mode that is directly involved in the torsional isomerization pathway and (c) photoisomerization in the S2(ππ*) state occurs via an inversion-assisted torsional pathway with a barrier of ~2 kcal mol⁻¹ separating the localized minimum of the S2(ππ*) state from a conical intersection with the ground state at similar energies as the barrier. Similarly, we have determined the structural changes occurring upon excitation to the strongly absorbing S3(ππ*) state. Comparison of the absorption and excitation spectra of this state have led to the conclusion that, at least from these spectra, no indication can be found for the importance of dynamical processes other than internal conversion to the S3(ππ*) state.

Elucidation of the details of the potential energy surfaces of the S1(nπ*) and S3(ππ*) states and their associated dynamics is important, since they ultimately determine the operation of azobenzene in photofunctional devices. In this respect, the present results could form the basis for a further rational design of novel molecular devices that use this important building block. At a more general level, the present studies have shown the important benefits high-resolution spectroscopy can offer for the study of (ultra)fast photodynamics. They should provide further impetus for similar studies on the vast amount of photoresponsive systems that so far have not been considered, because their lifetimes were considered to be prohibitively short.

Methods

Experimental details. In our studies, excitation spectra of cold isolated molecules were recorded by two-colour (1 + 1′) RE2PI spectroscopy on trans-azobenzene seeded in a supersonic expansion of neon. For this purpose, trans-azobenzene (Sigma-Aldrich) was placed in an oven that was heated up to 145 °C and expanded into vacuum with 2 bars of Ne via a pulsed nozzle with a 0.5-mm orifice (General Valve Inc., One system). Following the supersonic free jet expansion, a 2-mm conical skimmer skimmed the expansion prior to entering the ionization chamber. Ions were detected using a reflection time-of-flight mass spectrometer (R.M. Jordan Co.).

In the two-colour RE2PI experiments, a frequency-doubled dye laser (Sirah Precision Scan) pumped by a Nd:YAG laser (Spectra Physics Lab 190) operating at 30 Hz was used for excitation. Typical pulse energies were 2–3 mJ, but when necessary these were attenuated to avoid saturation. Ionization occurred by means of an ArF excimer laser (Neweks PSX-501) that typically produces pulses with energies of 5 mJ per pulse. The two lasers are introduced into the spectrometer in a counter-propagating fashion without further focusing. The pulsed nozzle and the laser systems were synchronized by a delay generator (Stanford Research Systems DG535). For the ultraviolet–ultraviolet depletion experiments, another laser system consisting of a frequency-doubled dye laser (Sirah Cobra-Stretch) pumped by a second Nd:YAG laser (Spectra Physics Lab 190) was employed. In these experiments, the depletion laser, which is scanned over the excitation energy of interest, was fired 100–200 ns prior to the lasers creating the (1 + 1′) RE2PI probe signal at a fixed excitation wavelength.

Computational details. Geometry optimization followed by calculation of the harmonic force field was performed for the S0 ground state and the S1(nπ*) and S3(ππ*) electronically excited states using time-dependent density functional theory. These calculations were performed at the B3LYP/6-31G(d,p) level utilizing the Gaussian09 suite of programs. The same program was employed for simulation of the vibrationally resolved excitation spectra within the Herzberg-Teller approximation. For comparison with experimental spectra, a scaling factor of 0.961 was used for the predicted frequencies.

References

1. Browne, W. R. & Feringa, B. L. Light switching of molecules on surfaces. Annu. Rev. Phys. Chem. 60, 407–428 (2009).
2. Fehrentz, T., Schönberger, M. & Trauner, D. Optochemical genetics. Angew. Chem. Int. Ed. 50, 12156–12182 (2011).
3. Bandara, H. M. D. & Burdette, S. C. Photoisomerization in different classes of azobenzene. Chem. Soc. Rev. 41, 1809–1823 (2012).
4. Patherm, B. K., Claridge, S. A., Zheng, Y. B. & Weiss, P. S. Molecule switches and motors on surfaces. Annu. Rev. Phys. Chem. 64, 605–630 (2013).
5. Rau, H. in Photochromism: Molecules and Systems Vol. 1 (eds Dürr, H. & Boudaiev-Laurent, H.J.) 165–192 (Elsevier, 1990).
6. Borotuls, P. & Monti, S. cis-trans photoisomerization of azobenzene. Solvent and triplet donors effects. J. Phys. Chem. 83, 648–652 (1979).
7. Siampiringue, N., Guyot, G., Monti, S. & Borotuls, P. The cis←→trans photoisomerization of azobenzene: an experimental re-examination. J. Photochem. 37, 185–188 (1987).
8. Rau, H. & Lueddecke, E. On the rotation-inversion controversy on photoisomerization of azobenzenes. Experimental proof of inversion. J. Am. Chem. Soc. 104, 1616–1620 (1982).
9. Schultz, T. et al. Mechanism and dynamics of azobenzene photoisomerization. J. Am. Chem. Soc. 125, 8098–8099 (2003).
10. Lednev, I. K., Ye, T.-Q., Hester, R. E. & Moore, J. N. Femtosecond time-resolved UV – visible absorption spectroscopy of trans-azobenzene in solution. J. Phys. Chem. 100, 13338–13341 (1996).
11. Nägge, T., Hoche, R., Zinth, W. & Wachtveitl, J. Femtosecond photoisomerization of cis-azobenzene. Chem. Phys. Lett. 372, 489–495 (1997).
12. Lednev, I. K. et al. Femtosecond time-resolved UV-visible absorption spectroscopy of trans-azobenzene: dependence on excitation wavelength. Chem. Phys. Lett. 290, 68–74 (1998).
13. Fujino, T. & Tahara, T. Picosecond time-resolved Raman study of trans-azobenzene. J. Phys. Chem. A 104, 4203–4210 (2000).
14. Satzger, H. et al. Fluorescence spectra of trans- and cis-azobenzene—emission from the Franck–Condond state. Chem. Phys. Lett. 372, 216–223 (2003).
15. Satzger, H., Root, C. & Braun, M. Excited-state dynamics of trans- and cis-azobenzene after UV excitation in the ππ* band. J. Phys. Chem. A 108, 6265–6271 (2004).
16. Chang, C.-W., Lu, Y.-C., Wang, T.-T. & Diau, E. W.-G. Photoisomerization dynamics of azobenzene in solution with S1 excitation: a femtosecond fluorescence anisotropy study. J. Am. Chem. Soc. 126, 10109–10114 (2004).
17. Stuart, C. M., Frontiera, R. R. & Mathies, R. A. Excited-state structure and dynamics of cis- and trans-azobenzene from resonance Raman intensity analysis. J. Phys. Chem. A 111, 12072–12080 (2007).
18. Chowdary, P. D. & Umapatthy, S. Ultrafast dynamics and photochemistry of ππ* excited trans-azobenzene—a comprehensive analysis of resonance Raman intensities. J. Raman Spectrosc. 39, 1538–1555 (2008).
19. Quirk, M. et al. Photoisomerization dynamics and pathways of trans- and cis-azobenzene in solution from broadband femtosecond spectroscopies and calculations. J. Phys. Chem. B 118, 8756–8771 (2014).
20. Ishikawa, T., Noro, T. & Shoda, T. Theoretical study on the photoisomerization of azobenzene. J. Phys. Chem. A 115, 7503–7512 (2001).
21. Cembran, A., Bernardi, F., Garavelli, M., Gagliardi, L. & Orlandi, G. On the mechanism of the cis → trans isomerization in the lowest electronic states of azobenzene: S0, S1, T1. J. Am. Chem. Soc. 126, 3234–3243 (2004).
22. Wang, Q & Diau, E. A new trans-cis photoisomerization mechanism of azobenzene on the S1(nπ*) surface. J. Phys. Chem. A 108, 950–956 (2004).
23. Gagliardi, L., Orlandi, G., Bernardi, F., Cembran, A. & Garavelli, M. A theoretical study of the lowest electronic states of azobenzene: the role of torsion coordinate in the cis-cis photoisomerization. Theor. Chem. Acc. 111, 363–372 (2004).
24. Conti, L., Garavelli, M. & Orlandi, G. The different photoisomerization efficiency of azobenzene in the lowest nπ* and ππ* singlets: the role of a phantom state. J. Am. Chem. Soc. 130, 5216–5230 (2008).
25. Yuan, S., Dou, Y., Wu, W., Hu, Y. & Zhao, J. Why does trans-azobenzene have a smaller isomerization yield for ππ* excitation than for nπ* excitation? J. Phys. Chem. A 112, 13326–13332 (2008).
26. Wang, L., Xu, W., Yi, C. & Wang, X. Isomerization and electronic relaxation of azobenzene after being excited to higher electronic states. J. Mol. Graph. Model. 27, 792–796 (2009).
27. Tiberio, G., Muccioli, L., Berardi, R. & Zannoni, C. How does the trans–cis photoisomerization of azobenzene take place in organic solvents? Chem. Rev. 111, 1016–1042 (2011).

28. Cusati, T., Granucci, G. & Persico, M. Photodynamics and time-resolved fluorescence of azobenzene in solution: a mixed quantum-classical simulation. J. Am. Chem. Soc. 133, 5109–5123 (2011).

29. Harabuchi, Y., Ishii, M., Nakayama, A., Noro, T. & Takeuchi, T. A multireference perturbation study of the NN stretching frequency of trans-azobenzene in nπ* excitation and an implication for the photoisomerization mechanism. J. Chem. Phys. 138, 064305 (2013).

30. Ferrante, C., Kensy, U. & Dick, B. Does diphenylacetylene (tolan) fluoresce from its second excited singlet state? Semiempirical MO calculations and fluorescence quantum yield measurements. J. Phys. Chem. 97, 13457–13463 (1993).

31. Heikal, A. A., Baskin, J. S., Ban˜ares, L. & Zewail, A. H. Structural effects on the trans–cis photoisomerization mechanism. J. Phys. Chem. A 101, 5017–5031 (2006).

32. Mu¨ller, C., Klo¨ppel-Riech, M., Schro¨der, F., Schroeder, J. & Troe, J. Fluorescence and REMPI spectroscopy of jet-cooled isolated 2-phenylindene in the S1 state. J. Phys. Chem. A 110, 5017–5031 (2006).

33. Tan, E. M. M., Amirjalayer, S., Bakker, B. H. & Buma, W. J. Excited state dynamics of photoactive yellow protein chromophores elucidated by high-resolution spectroscopy and ab initio calculations. Faraday Discuss. 163, 321–340 (2013).

34. Hochstrasser, R. M. & Lower, S. K. Polarization of the spectra of crystalline azobenzene-CHCl3: 1,108–1302 cm−1 of azobenzene in stilbene at 77° and 4.2°K in the region of the lowest n–π* transition. J. Chem. Phys. 36, 3505–3506 (1962).

35. Cusati, T., Granucci, G., Persico, M. & Spighi, G. Oscillator strength and vibronic coupling in 1,2,4,5-tetrafluorobenzene: the S1 spectrum. J. Chem. Phys. 105, 10492–10501 (1996).

36. Buma, W. J. & Zerbetto, F. Vibronic activity in trans,trans-1,3,5,7-octatetraene: the S1–S0 spectrum. J. Phys. Chem. A 103, 10492–10501 (1995).

37. Tan, E. M. M., Amirjalayer, S., Bakker, B. H. & Buma, W. J. Excited state dynamics of photoactive yellow protein chromophores elucidated by high-resolution spectroscopy and ab initio calculations. Faraday Discuss. 163, 321–340 (2013).

38. Zwier, J. M., Wiering, P. G., Brouwer, A. M., Bebelaar, D. & Buma, W. J. The lowest excited singlet states of cis,trans-1,3,5,7-octatetraene: the role of symmetry breaking and vibronic interactions. J. Phys. Chem. A 103, 2220–2226 (1998).

39. Doktorov, E. V., Malkin, I. A. & Man’ko, V. I. Dynamical symmetry of vibronic transitions in polyatomic molecules and the Franck-Condon principle. J. Mol. Spectrosc. 64, 302–326 (1977).

40. Orlandi, G. & Zerbetto, F. Vibronic coupling in 1,2,4,5-tetrafluorobenzene: the double-minimum potential of the butterfly motion (Q11) in the state S1. J. Phys. Chem. 91, 4238–4240 (1987).

41. Orlandi, G. & Zerbetto, F. Vibronic coupling in 1,2,4,5-tetrafluorobenzene: the double-minimum potential of the butterfly motion (Q11) in the state S1. J. Phys. Chem. 91, 4238–4240 (1987).

42. Orlandi, G. & Zerbetto, F. Vibronic coupling in 1,2,4,5-tetrafluorobenzene: the double-minimum potential of the butterfly motion (Q11) in the state S1. J. Phys. Chem. 91, 4238–4240 (1987).

43. Chevy, R., Offerle, R., Stoll, H. & Weigend, F. Electronic coupling in 1,2,4,5-tetrafluorobenzene: the double-minimum potential of the butterfly motion (Q11) in the state S1. J. Phys. Chem. 91, 4238–4240 (1987).

44. Dyck, R. H. & McClure, D. S. Ultraviolet spectra of stilbene, p-monohalogen stilbenes, and azobenzene and the trans to cis photoisomerization process. J. Chem. Phys. 36, 2326–2345 (1962).

45. Andersson, J.-Å., Petterson, R. & Tegnér, L. Flash photolysis experiments in the vapour phase at elevated temperatures I: spectra of azobenzene and the kinetics of its thermal cis-trans isomerization. J. Photochem. 20, 17–32 (1982).

46. Biancaland, A. et al. Direct observation of a 220 cm−1 structure in the lowest π–π* absorption band in the vapour phase of trans-azobenzene. Spectrochim. Acta A Mol. Biomol. Spectrosc. 55, 2883–2887 (1999).

47. Fujino, T., Azhantsev, S. Y. & Tahara, T. Femtosecond time-resolved fluorescence study of photoisomerization of trans-azobenzene. J. Phys. Chem. A 105, 8123–8129 (2001).

48. Frisch, M. J. et al. Gaussian 09, Revision A.01 (Gaussian Inc., 2009).

49. Tsuji, T., Takashima, H., Takeuchi, H., Ejawa, T. & Konaka, S. Molecular structure and torsional potential of trans-azobenzene. A gas electron diffraction study. J. Phys. Chem. A 105, 9347–9353 (2001).

Acknowledgements
This work was supported by The Netherlands Organization for Scientific Research (NWO). S.A. would like to thank the Deutsche Akademie der Naturforscher Leopoldina—German National Academy of Sciences for a Leopoldina research fellowship (grant number LPDS 2011-18).

Author contributions
E.M.M.T. designed and performed the experiments, performed the calculations and analysed the data. S.A. assisted with the experiments, calculations and their interpretation. S.S. and A.V. designed and performed initial experiments on the S2(ππ*) state that formed the basis for the experiments reported here. F.Z. helped with the interpretation of the experimental data. W.J.B. designed the experiments and supervised the experiments, calculations and data analysis. E.M.M.T. and W.J.B. co-wrote the manuscript with input from S.A. and F.Z.

Additional information
Supplementary Information accompanies this paper at http://www.nature.com/naturecommunications

Competing financial interests: The authors declare no competing financial interests.

Reprints and permission information is available online at http://npg.nature.com/reprintsandpermissions/

How to cite this article: Tan, E. M. M. et al. Fast photodynamics of azobenzene probed by scanning excited-state potential energy surfaces using slow spectroscopy. Nat. Commun. 6:5860 doi: 10.1038/ncomms6860 (2015).

This work is licensed under a Creative Commons Attribution 4.0 International License. The images or other third party material in this article are included in the article’s Creative Commons license, unless indicated otherwise in the credit line; if the material is not included under the Creative Commons license, users will need to obtain permission from the license holder to reproduce the material. To view a copy of this license, visit http://creativecommons.org/licenses/by/4.0/