Model of Global Permanent Ischemia in Conditions of Normothermia and Hypothermia

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ABSTRACT

We propose a model of global permanent ischemia on rats, obtained through decapitation of the animals, in which successive metabolic examinations have been conducted at different ischemia times up until a maximum of 2 hours. In order to verify the protective role of temperature’s reduction, metabolic examinations have been conducted at three different temperatures: normothermia (37ºC); moderate hypothermia (34ºC); and low hypothermia (31ºC). For each different temperature examined the values are the results of the average of 6 animals, this to limit the alterations deriving from the unusual behavior of single specimen, consequence of the inter-subject variability inside the same species. The reference values were obtained from the brain metabolic examination of 6 rats carried out immediately after the decapitation process. The metabolic inquiries carried out were focused on the analysis of the changes of the cellular energetic metabolism through the dosage of main energy molecules (ATP and GTP), of their intermediate catabolites (ADP, AMP, GDP, GMP) and of their final products (hypoxanthine, xanthine, uric acid, adenosine and inosine); on the mitochondrial alterations, through the note of the modifications of the nicotinic coenzymes entailed in the oxidative phosphorylation process (NAD and NADP) and on the action of the main mediators of the ischemic damage. From the analysis of the results presented above, in our experimental model the result is that in conditions of global ischemia and normothermia there is a progressive reduction of the main energetic metabolites

Introduction

Despite the huge amount of literature on cerebral metabolism and cerebral ischemia, a clear definition of ischemic cell death universally accepted is yet to be found. Furthermore, there is no unanimous agreement on the timing of neuronal cells death secondary to an ischemic insult. Several studies have shown that distinct neuronal populations have a different vulnerability to ischemia [1-3]. There are reports in the literature on how a 5-minute global ischemia in the gerbil cause delayed death in almost all pyramidal cells of CA1 (Ammon’s horn), without any effect detrimental in other neuronal populations, while 20 minutes of ischemia are sufficient to induce cell death in CA3 and do not create damage in the hippocampal dentate granule cells [4,5]. Other studies have also shown that the same threshold of tolerance of a neuronal population might be changed allowing recovery of the ability of protein synthesis or interrupting the dissolution process of intracellular organelles [6,7].

Alongside with the progresses in analyzing the pathogenesis behind the ischemic insult, neuroprotection mechanisms have become matter of interest. In particular studies on all those methods, pharmacological, physical or chemical that can reduce or delay the deleterious effects of the absence of blood flow were considered. Among those neuroprotective measures, hypothermia plays an extremely important role. Numerous studies have shown that even small reductions in brain temperature induced during
or after the ischemic insult, provide adequate neuroprotection [8-13]. Hypothermia has now assumed a prominent role in studies on ischemic brain and it is for this reason that a large part of this work will be devoted to the analysis of the effects of the reduction in brain temperature, even in light of our experience.

The model that we propose in this experimental study is a situation of global permanent ischemia on rats, obtained through the decapitation of the animals on which the successive metabolic examinations have been conducted at different ischemia times up until a maximum of 2 hours. In order to verify the protective role of the temperature's reduction along the ischemic process, the metabolic examinations have been conducted at three different temperatures: normothermia (37°C); moderate hypothermia (34°C); and low hypothermia (31°C). In order to reduce the variability of the analysis deriving from the unusual behavior of single specimen, consequence of the inter-subject variability inside the same species 6 animals were analyzed. The reference values were obtained from the brain metabolic examination of 6 rats carried out immediately after the decapitation process (time 0'). Metabolic analysis was focused on the sampling of the changes of the cellular energetic metabolism through the dosage of main energy molecules (ATP and GTP), of their intermediate catabolites (ADP, AMP, GDP, GMP) and of their final products (hypoxanthine, xanthine, uric acid, adenosine and inosine). Furthermore, analysis on the mitochondrial alterations, through the note of the modifications of the nicotinic coenzymes entailed in the oxidative phosphorylation process (NAD and NADP) and on the action of the main mediators of the ischemic damage, represented by the free radicals whose activity was indirectly measured through the modifications of the malondialdehyde levels (MDA), resulting from the action of those substances with the membrane's phospholipids, just like those of the main cellular antioxidant (ascorbic acid).

Materials and Methods

Preparation of the Isolated Rat Brain

Male adult rats were anesthetized through the intraperitoneal administration of Ketamine (300mg/kg of body weight) and, after the injection of 1000 UI of heparin in the caudal vena cava, were decapitated in a way to allow the reproduction of a total ischemic state, without the influence of collateral circles. The whole head of the animals was then put in special containers for the subsequent thermic treatment. Three groups of six specimen each were put in incubator, in controlled humidity conditions, at the constant temperature of 37°C for the following times: 30 minutes, 60 minutes, and 120 minutes. The same temporal kinetic was followed also for two more series of three groups of six specimen each put at the constant temperatures of 34 and 31°C respectively.

In order to provide a control group of brains, six anesthetized and decapitated animals were used and immediately processed for the biochemical examinations. At the end of each period of thermic treatment, the brains were immediately immersed in liquid nitrogen and processed to obtain a deproteinized tissular extract, to be used for the definition of the biochemical parameters selected through high-performance liquid chromatography (HPLC) technique. The brains were taken from the liquid nitrogen and immediately homogenized for 60 seconds in the presence of perchloric acid (HClO₄) 1.2 M cold, so to obtain the complete deproteinization of the specimen. The Homogenate of each specimen obtained in the above-mentioned way has been centrifuged for 15 minutes at 13000rpm (at 4°C); the supernatant was separated and stored at 4°C, while the pellet obtained was newly homogenized with HClO₄, 1.2M cold. After a further centrifugation (13000 rpm/10 minutes, 4°C), the new supernatant was added to the previous of each specimen and the whole was neutralized through the addition of K₂CO₃. Following a new centrifugation (13000 rpm/10 minutes, 4°C), the pellet of sunken salt was discarded and an equal volume of chloroform (1:1; v:v) was added to the neutralized tissular extract, in order to obtain the removal of all the non-water-soluble components present in the specimen. After having whirled the specimen for 60 seconds, a centrifugation of the specimen at 10000 rpm for 10 minutes at 4°C was carried out; the superior aqueous phase thus obtained was drawn (composed then exclusively by the aqueous supernatant free from any lipoid leftover) which was stored at -80°C until the time of the HPLC examination, before which the specimen was accurately filtered with a 0.45μM MILLIPORE-HV filter.

Biochemical Parameters of the Research

On the neutralized acid tissular extracts of rat brain, the following concentrations were evaluated through HPLC: the high energy of the phosphorylated compounds (ATP, ADP, AMP, GDP, GMP, UTP, UDP, UMP), of the nicotinic coenzymes (NAD and NADP), of the ox purine (hypoxanthine, xanthine, uric acid), of the nucleosides (adenosine and inosine), of the antioxidant ascorbic acid and of the malondialdehyde (MDA) as biochemical index of lipid peroxidation induced by the oxygen radicals (ROS) on the polyunsaturated fat acids of the phospholipids of the biological membranes.

HPLC Chromatographic Method of Examination

HPLC consists in a small modification of the previous ionic coupling methods perfected in our labs, during which the tetrabutylammonium is used as a coupling agent. A rate of the aqueous phase (100μl) will be examined through high performance liquid chromatography (HPLC) whose apparatus consist of a double pump system ThermoQuest Constametric 3500 (ThermoFinnigan Italia, Rodano, Milano, Italia) connected to an in series-diode spectrophotometric developer Spectra System UV6000LP (ThermoFinnigan Italia) which was set between the wave lengths of 190 and 330 nm. All this system is, in turn, connected to a PC for the acquisition and analysis of data, which uses a software system supplied by the same company that produces the HPLC system. For the analysis of the specimen an analytic chromatographic column Kromasil C-18, 250 mm x 4.6 mm was used, with 5μM
porins (Eka Chemicals, AB, Bohus, Sweden) provided with the specific preclum. The column was balanced with a mobile phase (A tampon) containing Hydroxide tetrabutylammonium 10 mM, KHPO₄ 10 mM, and methanol at 0.25%, with 7.00 pH. A gradient “with leaps” was carried out through a second tampon (B) containing hydroxide tetrabutylammonium 2.8 mM, KHPO₄ 100 mM, and methanol at 30% with 5.50 pH; the gradient used is the following: 15 minutes at 100% of A; 5 minutes at 90% of A; 5 minutes at 70% of A; 15 minutes at 63% of A; 15 minutes at 55%; 3 minutes at 50% of A; 32 minutes at 48% of A; 10 minutes at 28% of A; 15 minutes at 0% of A. The flux of the chromatographic runs was 1.2 ml/min and the temperature were maintained constant at 23°C, through a specific thermostatic system. The separation of the compounds in the experiment, which are identified through the confrontation of runs of ultra-pure standards both of the respective retention times and of the absorbing spectrums, has permitted the calculus of the concentrations carried out at the specific wave lengths of each substance.

**Statistical Analysis of Data**

The results of the different experiments were analyzed with specific statistical tests as the two-tailed Student test for non-coupled specimen.

**Results**

The results emerged from the experiments refer to the rat groups that underwent the model of global ischemia at the temperature of 37, 34 and 31°C. The values represent the average of the groups of 6 rats at different observation times (30'-60'-120'). The values reported for time 0' are also the average of 6 animals that underwent at the metabolic investigations immediately after the decapitation process; they then represent the benchmark controls for each group. Room temperature is the value to be considered in this case. Detailed statistical analysis has been carried out only for the parameters correlated to the main cellular cycles. At 37°C, as we move away from the beginning of the ischemia, a progressive reduction of the energetic metabolites (Figure 1; Tables 1-5) can be observed. The data is confirmed by the parallel increase of those that can be considered the intermediate catabolites of the main cellular energetic molecules (ADP, AMP, GDP, GMP) and of their final catabolic products (hypoxanthine, xanthine, and adenosine). Also, the uric acid presents a net increment at 60' and 120' in comparison to the concentrations of the start; this substance is the final product of the action of the xanthine oxidase on the ox purines, described above in relation to the ways of formation of the free radicals. In turn, the gradual increase of MDA (Figure 2), index of the action of the free radicals on the membrane phospholipids, seems to perfectly correlate with the progressive reduction of the Ascorbic Acid which represents the main cellular antioxidant (Figure 3). Finally, the reduction of the concentrations of the nicotinic coenzymes (NAD and NADP) is index of the progressive alteration of the mitochondrial function, since the lower availability of these electron acceptors entails the gradual block of the mitochondrial respiratory chain and thus of the cellular oxidative phosphorylation. It is important to notice that, following the extraction of the specimen in perchloric acid, there is the total transformation of the compounds reduced in the correspondent oxidative form. Hence, in the particular case of the nicotinic coenzymes it is indicated with the acronym NAD (or NADP) the total concentration of NAD+NADH (or NADP + NADPH). It is interesting to note that the reduction of NAD levels (Figure 4) could be correlated to the parallel increase of the concentrations of ADP ribose (Figure 5) which, in turn, is the expression of the activity of the poly (ADP-ribose) polymerase (PARP) system. The correlations of the two occurrences will be examined in the discussion phase.

**Table 1:** ATP levels at different observation times and temperature.

| Time/Temperature | 31°C  | 34°C  | 37°C  |
|------------------|-------|-------|-------|
| 0                | 1571.32 | 1571.32 | 1571.32 |
| 30               | 1366.51 | 1253.64 | 1016.27 |
| 60               | 902.85  | 825.83  | 765.86  |
| 120              | 833.93  | 749.88  | 694.64  |

**Table 2:** MDA levels at different observation times and temperature.

| Time/Temperature | 31°C | 34°C | 37°C |
|------------------|------|------|------|
| 0                | 2.06 | 2.06 | 2.06 |
| 30               | 13.62| 16.00| 15.04|
| 60               | 18.33| 21.92| 21.15|
| 120              | 29.63| 36.46| 38.80|

**Table 3:** Ascorbic acid levels at different observation times and temperature.

| Time/Temperature | 31°C | 34°C | 37°C |
|------------------|------|------|------|
| 0                | 1778.20 | 1778.20 | 1778.20 |
| 30               | 1303.10 | 1167.90 | 1059.50 |
| 60               | 997.69  | 942.62  | 783.35  |
| 120              | 842.18  | 784.96  | 747.75  |

**Table 4:** NAD levels at different observation times and temperature.

| Time/Temperature | 31°C | 34°C | 37°C |
|------------------|------|------|------|
| 0                | 310.25 | 310.25 | 310.25 |
| 30               | 291.44 | 256.56 | 247.24 |
| 60               | 237.38 | 210.00 | 186.00 |
| 120              | 161.76 | 141.12 | 136.75 |

**Table 5:** ADP levels at different observation times and temperature.

| Time/Temperature | 31°C | 34°C | 37°C |
|------------------|------|------|------|
| 0                | 44.08 | 44.08 | 44.08 |
| 30               | 65.85 | 67.92 | 91.92 |
| 60               | 91.42 | 99.54 | 141.5 |
| 120              | 107.38| 111.11| 155.90|
Figure 1: ATP levels at different observation times and temperature.

| Time/Temperature | 31°C  | 34°C  | 37°C  |
|------------------|-------|-------|-------|
| 0                | 1571.32 | 1571.32 | 1571.32 |
| 30               | 1366.51 | 1253.64 | 1016.27 |
| 60               | 982.85  | 825.83  | 765.86  |
| 120              | 833.93  | 749.88  | 694.64  |

Figure 2.

* significantly different from zero time (p < 0.05)
** significantly different from other groups at the same time of observation (p < 0.01)
* significantly different from only one of the groups at the same time observation (p < 0.01)

Figure 3.

* significantly different from zero time (p < 0.05)
In conditions of low (34°C) and moderate (31°C) hypothermia, we note the occurrence of a beginning (30’) slowdown of the ATP depletion (Figure 1), the only statistically significant value compared to the control (time 0’) is indeed the one of the group of animals at 37°C (p<0.01). After 60 minutes, all the groups present statistically different values compared to the benchmarks, index of a significant reduction of the ATP concentrations; nevertheless, the group at 31°C shows a higher conservation of the energetic metabolite which results statistically meaningful compared to the other groups for the same observation time (p<0.05). Finally, after 120 minutes of ischemia there is no observation of significant differences between the three examined groups and obviously this confirms the self-evident reduction of the ATP concentration compared to the benchmarking values. The increase of the final catabolic products ATP and GTP, represented by the sum of the oxypurines (hypoxanthine, xanthine, and uric acid) and by the sum of the nucleosides (adenosine and inosines) mirrors the same trend, resulting only modestly inferior for the groups of hypothermic rats.
and it is therefore legitimate to expect that, after a longer ischemia compromised and the tissue irreversibly damaged. It is obvious graph, beyond which the energetic metabolism could be irreparably coincides with the impossibility of pointing out a cut-off value in the remains (Figure 1). The absence of a real fall of the ATP levels after 2 hours a certain value of energetic metabolites still available reducing gradually as time passes from the ischemic event, but even of about 90% of the cellular reserves [14-18]. Such a net reduction concentrations which would happen in the span of a few minutes the ATP levels is surely not surprising. Nevertheless, many works in progressive reduction of the main energetic metabolites. The fall in the of the hypoxanthine's metabolism; in this case the reaction of Xanthine-Dehydrogenase (XD), that is equally responsible for water (H\textsubscript{2}O\textsuperscript{−}) and superoxide (O\textsubscript{2}\textsuperscript{−}). Another enzyme exists in the cerebral cellules, the Xanthine-Dehydrogenase (XD), that is equally responsible for the of the hypoxanthine's metabolism; in this case the reaction of dehydrogenization doesn't lead to the production of free radicals. The XO derives from the proteolytic divage of the XD which is facilitated, in conditions of ischemia, by the presence of high concentrations of calcium which activate the proteases responsible for the transformation; in such a way the production of free radicals.

**Discussion**

From the analysis of the results it can be suggested that in conditions of global ischemia and normothermia there is a progressive reduction of the main energetic metabolites. The fall in the ATP levels is surely not surprising. Nevertheless, many works in the literature refer to a rapid depletion of the energetic metabolite concentrations which would happen in the span of a few minutes from the beginning of the ischemia and would provoke the reduction of about 90% of the cellular reserves [14-18]. Such a net reduction is not at all documented in this study. The ATP in fact seems to be reducing gradually as time passes from the ischemic event, but even after 2 hours a certain value of energetic metabolites still available remains (Figure 1). The absence of a real fall of the ATP levels coincides with the impossibility of pointing out a cut-off value in the graph, beyond which the energetic metabolism could be irreparably compromised and the tissue irreversibly damaged. It is obvious that the considered temporal window frame is anyhow limited and it is therefore legitimate to expect that, after a longer ischemia time, a final depletion of the energetic cellular reserves will present itself; nevertheless, after two hours of global ischemia, it remains surprising to still observe the persistence of a significant quota of ATP. If the reduction of the ATP levels is confronted at the different temperatures examined in the research, it can be observed that the reduction of the cerebral temperature doesn't seem able to preserve the energetic metabolism. Actually, after an initial slowdown of the ATP depletion, observable in the hypothermic rats after 30 minutes of ischemia, the following reduction of the energetic levels doesn't differ in a substantial way among the three groups. In other parts of this dissertation it was discussed about the fact that the supposed capacity of the hypothermia to induce, in ischemia conditions, a slowdown of the fall of the ATP values, probable consequence of a reduction of the demand for oxygen and of a global inhibition of the cerebral metabolism, is non a unanimously recognized data [19]. Some authors in fact did not observe any modification in the ATP levels in conditions of hypothermia compared to the normothermic controls after situations of ischemia. So, as it was documented an initial protective effect induced by the temperature reduction after 30 minutes from the beginning of the ischemia it appears that by widening the temporal window frame of observation any protective role linked to the hypothermia seems irremediably lost. The data emerged from our research seems therefore in line to what stated by earlier experimental evidences in which it was observed a preservation effect of the energetic metabolism from the reduction of the cerebral temperature only in case of short length ischemic distress [20]. After 60 minutes of ischemia in our experimental model, the group of animals at 31°C seems to tend towards a higher conservation of the energetic levels, differentiating in a statistically significant manner from the other examined groups. This data could suggest that only the more marked reductions of the cerebral temperature would allow the preservation in time of that initial protective effect.

The increase, in conditions of ischemia, of the concentration of the final products of the catabolism of ATP, and in particular of the oxypurines (xanthine, hypoxanthine and uric acid), was confirmed in our experimental model. The higher availability of these compounds is at the base of the stimulation of the biochemical way that leads to the formation of free radicals by the xanthine-oxidase enzyme (XO), through the reaction reported below [21]:

\[
\text{Hypoxanthine} + O_2 \rightarrow O_2\bullet + H_2O_2 + \text{UricAcid}
\]

The results of the oxidation reaction are represented by the superoxide (O\textsubscript{2}\textsuperscript{−}) and by the peroxide of hydrogen or oxygenated water (H\textsubscript{2}O\textsubscript{2}). Another enzyme exists in the cerebral cellules, the Xanthine-Dehydrogenase (XD), that is equally responsible for the of the hypoxanthine's metabolism; in this case the reaction of dehydrogenization doesn't lead to the production of free radicals. The XO derives from the proteolytic divage of the XD which is facilitated, in conditions of ischemia, by the presence of high concentrations of calcium which activate the proteases responsible for the transformation; in such a way the production of free radicals.
is consequently favored. It is important to note that, as it is inferred by the reported reaction, the formation of free radicals is a process that requires the availability of oxygen. In the present experimental model of global ischemia such availability is a function of a certain quantity of oxygen leftover at the tissue level and of the quota that the same tissue can acquire from the atmosphere.

This allows, albeit more controlled, for the production of those toxic substances. In our opinion, this is a particularly important concept because it reafirms what is many times stated in other parts of this dissertation: the production of free radicals is a process typical of the reperfusion period that is nevertheless in part influenced by mechanisms that begin to activate themselves before the restoration of the hematic flux. Also in vivo it is actually possible that one quota of leftover oxygen is available at the cellular level; furthermore, the data correlates perfectly with the models of temporary focal ischemia, in which the oxygen can spread through diffusion towards the infarct areas from the adjacent zones still perfused. In such conditions an increase of the formation of free radicals happens, even before the occlusion at the base of the situation of ischemia is removed [22,23]. Among the different ways of formation of the oxygen derivate, the one that plays an important role in the phase of ischemia in the strict sense is the reaction mediated by the XO, while the others activate only at a later stage when the hematic flux is reestablished [24,25].

As evidence of what has been stated, in the present model we observe a progressive raise of the concentrations of uric acid which represents the final product of the reaction previously described. Considering then the trend of the MDA levels (Figure 2), expression of the action of the free radicals on the membrane phospholipids, in conditions of normothermia, it can be noted that the increase of the concentrations, while present and clear, is not nearly comparable to the burst of activity that the oxygen derivate normally show at the moment of reperfusion, further evidence of the fact that only a minimal quantity of these substances was able to be formed. The most significant data is the one that differs from the analysis of the MDA in hypothermia conditions. Both at 34 and 31°C, at a distance of 30 and 60 minutes from the beginning of the ischemia, significant modifications of the MDA concentrations do not stand out, if compared with the ones reached for the equal observation times at 37°C. After 2 hours of ischemia the group of animals at 31°C is the only one to present a lower increment of MDA levels compared to the rats at 37°C.

These data do not contradict the by now confirmed capacity of the reduction of cerebral temperature to reduce the formation of free radicals in conditions of ischemia, but simply implicitly imply a different mechanism of action. The hypothermia is in fact essentially able to contrast the activity of the NOS and most of all of the inducible isofrom, which is mainly contained at the microglial cellular level, which, in turn, are activated by the inflammation and secondary damage processes that are carried out in the phase of reperfusion. From the research of the Hyung Soo Han group [24,25] previously described, it has emerged that the maximum peak of formation of the two isoforms of the NOS shows up at the distance of 24 hours from the ischemic event. It is true that from that research it has also emerged that the intraischemic hypothermia (33°C) was able to reduce the activity and cellular expression of the iNOS also at a distance of only 2 hours from the beginning of the ischemic event, but is also true that the model used in that case was one of a temporary focal ischemia and that the sample taken after 2 hours coincided with the restoring of the hematic flux. In a moment of permanent global ischemia obtained through decapitation it is unlikely to imagine the possibility that a situation of secondary inflammation might occur; the only way of formation of free radicals remains then the one described. In such conditions even a moderate hypothermia (31°C) results ineffective in contrasting the formation of dangerous derivate of oxygen, since it is not able to intervene in a significant manner on the reaction catalyzed by the XO.

In this experimental model, in conditions of normothermia, a progressive reduction of the concentrations of the nicotinic coenzymes is observed. This phenomenon is expression of the mitochondrial damage that happens following the ischemic event: the minor availability of the NAD and NADP, that are important electron acceptors, corresponds at a gradual block of the mitochondrial respiratory chain and consequently of the oxidative phosphorylation [26]. Such a process seems to be the result of the previously described mitochondrial alterations that occur in the first hours after the ischemic insult and that see involved, as a main responsible, the intracellular calcium [27]. Nevertheless, from the analysis of the results of this research emerges also another data that could be correlated to the reduction of the levels of the nicotinic coenzymes: the progressive increase of the quota of ADP-ribosé. The ribosylation of the ADP following the ischemic event is a process mediated almost exclusively by the action of the PARP system [28,29].

These cellular enzymes are activated through only one known system, represented by the single filament breaking of the DNA chain [30]. The alterations on account of the cellular nucleic acids are the result of the free radical actions: the peroxyxirite first and foremost, but also the hydroxyl radical and the superoxide are the compounds mainly involved [31]. For that reason, both the alterations at the DNA level both the subsequent activation of the PARP system are generally phenomena typical of situations of focal and temporary ischemia, showing up at the moment of the restoring of the hematic flux and of the supply of oxygen. Nevertheless, in our experimental model the presence of a certain quota of activity by the free radicals is indisputable, as shown by the increase, even if contained, of the MDA values. There are no reasons to thing that, in the same manner in which a damage action is carried out toward the phospholipid membranes, an alteration could not be carried out, mediated by the radicals, at cellular DNA level, able to activate the enzymatic system of the poli-ADP-ribose-polymerase. This theory,
without an objective evidence of the presence of DNA alterations and without the use of specific inhibitors for the enzymatic systems in question, cannot be considered completely well grounded. Nevertheless, the registered raise of the ADP-ribose levels would seem to support it. If so should be, the progressive reduction of the NAD concentrations could be interpreted also in light of a direct action by the PARP system on the nicotinic coenzyme, since these enzymes are able to provoke a cleavage of the NAD, becoming the main responsible of its depletion in conditions of focal ischemia.

Furthermore, this model would demonstrate that an activation of the PARP system, maybe even partial, can happen even while in absence of a restoring of the hematic flux. Future research would anyway be necessary to confirm this interesting theory. The analysis of the trend of the ADP-ribose values and of the nicotinic coenzymes in conditions of hypothermia provides important elements of reflection. It can be observed that, both at 31 and at 34°C after 2 hours of ischemia a net fall of the ADP-ribose values associated to the keeping, in the case of the group of rats at 31°C, or higher concentrations of NAD compared to the normothermic controls. This data could indicate a protective effect induced by the reduction of the cerebral temperature that would be exercised directly towards the activity of the PARP system. As a matter of fact, since significant modifications of the free radicals activity between the groups of hypothermic animals and the one at 37°C were not recorded in the present experimental model, the reduced concentrations of ADP-ribose documentable at 31 and also at 34°C, cannot be interpreted on the base of a minor activation of the enzymatic system subsequent to a lower damaging action at the nuclear level from the oxygen derivate; it is therefore more likely that the hypothermia is capable to directly antagonize the action of the enzymes, once that these are normally activated.

It is fair to underline that at 34°C, together with a reduction of the ADP-ribose concentrations mainly comparable to the one that happens at 31°C, a significant difference of the NAD levels compared to those recorded at 37°C is not evident; the data should not come as a surprise, since it is to be remembered that in conditions of ischemia other mechanism exist that can provoke the reduction of the NAD levels, different from the cleavage action exercised by the PARP system. It is sufficient to think that the synthesis of the nicotinic coenzyme is an ATP dependent process and that it is carried out partly inside the mitochondria that begin to present marked alterations of their structure and function already after an hour of the beginning of the ischemia.

Moreover, as it has been demonstrated by other researches [7,34], the NADH molecule, just like the NADPH one, are another preferred target of the hydroxyl radicals. The interaction of the OH• with the NADH and NADPH provokes both their oxidation at NAD+ and NADP+ and the production, on their expense, of ADP-ribose and nicotinamide. This reaction would justify on one hand the drop of the NAD levels, and on the other the increase of the ADP-ribose concentrations. Moreover, it is important to underline that the action of the hydroxyl radicals and the subsequent depletion of the pool of nicotinic coenzymes jeopardize the correct functioning both of the cellular energetic metabolism and of the oxide-reductive one, making more complicated the recovery of the cellule in the possible following phase of reperfusion.

Conclusion

Hypothermia is indeed one of the most important neuroprotective measures and through this animal model it was possible to demonstrate how even after 2 hours from an ischemic event a certain value of energetic metabolites still available remains. In addiction it seems that this protective effect might be due to the activity of the PARP system. Finally, metabolic interruption seems to be not an instantaneous process but rather something occurring in a longer period of at least 2 hours.

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None

Declarations of Interest

The authors declare that they have no conflict of interest.

Ethical Approval

All applicable international, national, and/or institutional guidelines for the care and use of animals were followed.

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