Comparative Study of Some Alternatives for the Reuse of Calcareous Mud in Egypt

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Abstract

The chemical analysis of the calcareous mud (CM) which is produced as a by-product from the Egyptian sugar beet companies was the corner stone behind the idea of this research. From the chemical analysis, it was found that the CM consists mainly of CaCO₃, MgCO₃, and about from 10 to 15% organic matter. The organic matter present in the CM is considered by all means as an added value, especially when the CM is going to be used in pyro-industrial applications. In this paper, three different alternatives will be investigated in order to choose the best of them.

The first alternative was replacing the limestone (LS) used in the sintering of El-Bahareya iron ores in the Egyptian Iron and Steel Company (Hadisolb) by the CM. In this respect, several pilot scale experiments were performed in Hadisolb. The performed experiments covered replacements of LS by the CM from zero to 100%, with an increment of 10% each time. Chemical analyses and drum tests were performed for each run of experiments. Taking into considerations both the productivity and the sinter quality, the results proved that at least 10% reduction in coke consumption can be achieved, due to the replacement of LS in the sintering process by the CM.

The second alternative is using the CM in the Charge Mix of the Calcination Furnace in cement companies. The performed experiments in Alexandria Portland Cement Company (APCC) proved that although the use of the CM in the mix is technically useful, but is not economic.

Aside from these two alternatives (pyro-industrial applications), a third alternative based on the advantage of offering CM for free, is used as a back-filling material (BFM) for the underground cables carrying the electric current. It is well known that the current carrying capacities of the cables are highly affected by the thermal resistivity of the BFM. The results of using the CM as a BFM for the underground cables compared with sand which is used nowadays by the current transmission companies in Egypt indicated that the current carrying capacity for cables of the same material and diameter was highly increased by using the CM as a BFM instead of sand.

Keywords

Calcareous mud, Sintering, Calcination furnace, Back-filling material, Thermal resistivity, Current carrying capacity
Introduction

There are eight sugar beet companies in Egypt now. Beside sugar they produce several by-products, including the calcareous mud (CM), which is the only unused product, considered as a waste material, and is accumulated year after year causing a lot of environmental and economic problems. As a matter of fact, the amount of the CM is about 8-12 w/o of the processed beet, which means that the total annual production of the CM is estimated to be about 800,000 tons. It may be worthy to mention here that in Europe the CM is reused in several aspects, such as an additive to animal feed, as a filling material in some industrial products (e.g., rubber, plastic, paper, in the building industry, and in water treatment). However, in Egypt the CM is still unused, in spite of the fact that the sugar beet companies offer to give it free of charge.

Far from the pyro-industrial applications, and taking an advantage of giving the CM free of charge, it may be worthy to suggest to the electric current transmission companies to use it as a back-filling material (BFM) for the underground cables instead of sand.

Materials and Experiments

First alternative: Iron ore sintering in (Hadisolb) [1]

In many cases, sintering is considered as an essential step in ore preparation for the blast furnace operation. Sintering may be defined as the agglomeration of fine particles into a strong porous mass. The process is carried out by heating the sintering charge at a temperature approaching its fusion. The sintering charge usually includes, beside the iron ores, the required amounts of LS as a flux, coke and water to produce sinter of a specific composition and quality [2,3].

The Egyptian Iron & Steel Company (Hadisolb) is the only company among all the other steel making companies in Egypt which adopts blast furnace/sintering route. The blast furnaces in Hadisolb use self-fluxed sinter in its burden. Self-fluxed sinter, as was mentioned before, requires an iron ore, a fluxing material such as LS and/or CM, coke and water.

Material

Iron ore: Iron ore deposits at El-Gedida is the only iron ore used now in Hadisolb. This ore is located in three different localities. At El-Gedida mines, ores from the three localities are extracted and blended according to predetermined program to meet the previously agreed upon composition, which is Fe > 51%, Cl < 0.6%, MnO < 2.4%, SiO₂ < 8%, CaO > 0.5%, and Al₂O₃ < 2% [4].

Limestone (LS): The LS used as a flux in the pilot-scale experiments is brought from Bani-Khalid quarries nearby Samalut. The quarry is owned by Hadisolb, and is one of the highest quality in Egypt. The LS was crushed and sieved up to 3 mm.

Calcereous mud (CM): In this investigation, a scientific approach for the reuse of the CM is adopted. This approach is based on the chemical analysis of the CM. The chemical analysis for a representative sample of CM was performed using XRF - 9800 ARL technique, and is given in Table 1 [5]. The chemical analysis given in Table 1 shows that the CM is composed mainly of CaCO₃, MgCO₃ and a significant amount of organic matter varies between 10-15% expressed as organic CO₂, and its value depends on the beet juice composition. The L.O.I. is due to evaporation of water and organic matter. We may consider that the 800,000 tons of CM - assuming only 10% organic matter - will contain about 80,000 tons of a fuel, accordingly, it was suggested to reuse the CM in the sinter plant of the iron and steel company and/or cement companies as a substitute for the LS used in any of them.

The idea of replacing LS by CM in the sintering process is based on the chemical composition of both materials. The chemical analysis for representative samples from Bani-Khaled quarries LS used as a flux material in the sintering plant of Hadisolb and that of CM from Dakahleya Sugar Beet Companies is given in Table 2.

Experiments: Hadisolb is the only company in Egypt, which has sintering plant. In this plant the iron bearing material goes through its final step of

| %CaCO₃ | %MgCO₃ | Elements% | Organic CO₂ % |
|-------|-------|-----------|--------------|
| %CaO | %CO₂ | %MgO | %CO₂ | %Al₂O₃ | %Fe₂O₃ | %P₂O₅ | %SiO₂ | %SO₃ | L.O.I. |
| 36.19 | 28.5 | 5.7 | 8.3 | 0.24 | 0.17 | 1.3 | 4 | 0.56 | 50.09 | ~15 |

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preparation since the blast furnaces, are fed by the iron bearing material in the form of sinter.

In this investigation, several pilot scale experiments were performed in the sintering pilot plant in Hadisolb. The sintering process requires Coke, Limestone and Moisture beside the iron bearing material. Sintering of different mixtures with various CM content (ranging from 0 to 100% in steps of 10%) is conducted in a pilot scale unit of 20 kg capacity within the sintering plant of Hadisolb, as shown in Figure 1 [6].

Second alternative: Charge mix of the calcination furnace in Alexandria Portland Cement Company (APCC)

The performed experiments in APCC proved that although the replacement of LS by the CM in the calcination furnace is technically feasible, it was found that it is not economic since the cement factories in Egypt have their own nearby LS quarries.

We estimate that for the Alexandria plant capacity 1,400,000-ton clinker/year X 3600 MJ/ton, approximately 5,040,000 GJ/year, the expected cost is about 32.4 EGP/GJ, so the total fuel cost is about 16.2 MEGP/year, and the saving is expected to be 0.6%, which means the saving would be around 979,500 EGP/year.

Table 2: Chemical analysis for representative samples from Bani-Khaled quarries LS flux used in the sintering plant of Hadisolb compared with CM from Dakahleya Sugar Beet Company.

| Compound | Limestone (LS) | Calcareous mud (CM) |
|----------|----------------|---------------------|
| Moisture | 0.42           | ~2                  |
| CaO      | 54.71          | 36.19               |
| Al₂O₃    | 0.06           | 0.24                |
| Fe₂O₃    | 0.12           | 0.17                |
| MgO      | 0.5            | 5.7                 |
| SiO₂     | 0.17           | 4.0                 |
| SO₃      | 0.17           | 0.56                |
| P        | 0.008          | 0.01                |
| Organic CO₂| --            | ~15                 |
| L.O.I.   | 43.78          | 50.09               |

Figure 1: Pilot scale sintering unit.
Third alternative: The use of CM as a back-filling material (BFM) for underground cables [7]

Three trials may be undertaken to lower the cost of transmitting electric current by means of underground power cables. The first trial is by selecting cables made from metals of higher electrical conductivity such as copper. The second is by using cables of larger diameters, and the third is by using a BFM of lower thermal resistivity. It is clear that the third one is less expensive compared to the other two, although the third one is less expensive it will lead also to an increase in the current carrying capacities of the cables. As a matter of fact, underground power cables are highly affected by the thermal resistivity of the BFM used as surroundings for the cables. It was found that the current carrying capacity increases as the thermal resistivity of the BFM decreases, and vice-versa. In Europe, sand is commonly used as a BFM for the underground power cables. The thermal resistivity of wet sand is relatively low, it is about 55 k°.cm/W, however it gradually increases as sand becomes dry, it becomes, 75,105 and 396 k°.cm/W when water content in sand is 20, 10 and ≈ 0%. In countries of dry climate such as Egypt and most of the Arab countries, the thermal resistivity of sand is very close to 396 k°.cm/W, and thus sand should not be recommended to be used as a BFM for the underground power cables in such countries. In this investigation a trial was made to urge the decision makers in the electric current transmission company in Egypt to use another local material of lower thermal resistivity as a BFM than that of sand [8-10].

Table 3: Average chemical analysis for different sinters with different CM%.

| No | CM% | Total Fe% | FeO% | Fe2O3% | SiO2% | CaO% | MgO% |
|----|-----|-----------|------|--------|-------|------|------|
| 1  | 0   | 52.20     | 13.60| 59.35  | 8.82  | 9.69 | 1.120|
| 2  | 10  | 52.15     | 13.6 | 59.2   | 8.80  | 9.55 | 1.200|
| 3  | 20  | 52.09     | 14.18| 58.62  | 8.44  | 9.10 | 1.36 |
| 4  | 30  | 52.05     | 14.80| 57.50  | 8.30  | 8.95 | 1.68 |
| 5  | 40  | 52.05     | 16.68| 55.57  | 8.06  | 8.88 | 2.20 |
| 6  | 50  | 52.00     | 15.56| 57.29  | 7.43  | 7.83 | 2.35 |
| 7  | 60  | 52.04     | 13.92| 58.41  | 7.86  | 7.51 | 2.60 |
| 8  | 70  | 52.00     | 16.87| 55.35  | 7.78  | 7.4  | 2.80 |
| 9  | 80  | 51.82     | 14.29| 58.67  | 7.60  | 7.30 | 3.10 |
| 10 | 90  | 51.78     | 15.87| 55.75  | 7.48  | 7.27 | 3.20 |
| 11 | 100 | 51.57     | 15.12| 56.88  | 7.20  | 7.20 | 3.30 |

Results and Discussion

In order to investigate the effect of replacement of LS by CM in the sintering process, several sintering experiments were performed, using different percentages of replacements. Representative samples were taken from each three similar experiments (of the same replacement percent), for chemical analysis. Table 3 and Figure 2 show the average chemical analysis for different sinters (3 samples for each composition) as
a function of CM% in the sample. From Table 3, it is noticed that the CaO% in the sinter decreases and the MgO% increases as the CM percent increases, which is in favor for the blast furnace operation, since the Mg increases the fluidity of the slag, i.e. decreases the viscosity, which enhances desulfurization. CaO is decreasing as the CM increases because the CM contains less CaO than LS, as was shown in Figure 1.

Table 4 and Figure 3 show the savings in the coke consumption in the sintering process as the result of replacing LS by CM. The drum test is a standard test in the Iron & Steel industry used to measure the sizes in the sample. It can be concluded from Table 4 that the total saving of coke in the sintering plant of Hadisolb may reach a value of about 15%, in addition to the saved amount of LS replaced by the CM.

Conclusions

Chemical analysis for the representative sample of CM showed that it contains a significant amount of organic matter which varies between 10-15% and this organic matter behaves as a fuel. Pilot scale experiments proved the feasibility of using CM in the sintering process in the Egyptian Iron and Steel Company (Hadisolb), and this leads to a reduction in coke consumption up to 15%.

Using CM in the calcination furnace at APCC, although found to be technically possible, but not economic.

The use of the CM as a B.F material for the underground cables led to an increase in the current carrying capacity of the underground cables due to its lower thermal resistivity compared to sand.

![Figure 2: Average chemical analysis for different sinters vs. CM%](image-url)

Table 4: Savings in coke consumption for different sinters with different CM%.

| No | CM (%) | Saving in coke (%) | Drum test (%) |
|----|--------|--------------------|---------------|
| 1  | 0 (100% LS) | 0 | 12.0 |
| 2  | 10 (90% LS) | 2 | 12.3 |
| 3  | 20 (80% LS) | 4 | 12.5 |
| 4  | 30 (70% LS) | 6 | 12.8 |
| 5  | 40 (60% LS) | 8 | 13.0 |
| 6  | 50 (50% LS) | 9 | 13.2 |
| 7  | 60 (40% LS) | 10 | 13.5 |
| 8  | 70 (30% LS) | 11 | 13.8 |
| 9  | 80 (20% LS) | 12 | 14.0 |
| 10 | 90 (10% LS) | 14 | 14.2 |
| 11 | 100 (0% LS) | 15 | 14.5 |
Results of Sintering using Calcareous Mud

Figure 3: Savings in the coke consumption in the sintering process as the result of replacing LS by CM.s.

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