Consciousness Energy Healing Treatment: Impact on the Physicochemical and Thermal Characteristics of Folic Acid

Gopal Nayak¹, Mahendra Kumar Trivedi¹, Alice Branton¹, Dahryn Trivedi¹, Snehasis Jana², *

¹Trivedi Global, Inc., Henderson, USA.
²Trivedi Science Research Laboratory Pvt. Ltd., Bhopal, India.

Abstract

Folic acid is essential in the body for the production of DNA and also plays various vital functions within the body. This study on folic acid was done to analyze the results of the Trivedi Effect® on its physicochemical and thermal properties with the help of analytical techniques. The method involves dividing the test sample into two parts. The first part was termed as a control sample, and no treatment was given to it; while the second part received the Biofield Energy Treatment remotely by a renowned Biofield Energy Healer, Gopal Nayak and termed as the treated sample. The study revealed that the particle sizes reduced by 7.01% (d₁₀), 6.53% (d₅₀), 8.37% (d₉₀), and 15.99% {D(4,3)}, thus increased the surface area by 1.72%, in the treated sample compared with the control sample. The PXRD data showed changes in the peak intensities and crystallite sizes ranging from -36.81% to 113.41% and -15.79% to 318%, respectively, thus increased in the average crystallite size by 22.55% of the treated sample compared with the control sample. The TGA analysis of the treated sample revealed a 3.94% increase in total weight loss that resulted in the remarkable reduction in the residual amount by 14.99%, in comparison to the untreated sample. The latent heat of fusion (ΔHfusion) and latent heat of decomposition (ΔHdegradation) were significantly altered by -10.33% and 10.46%, respectively in the treated sample compared to the untreated sample. The study denoted that the Biofield Energy Treatment is a novel approach that could be used to develop some polymorph of folic acid and that might improve its solubility, dissolution, and bioavailability in comparison to the control sample. Hence, the treated folic acid in the nutraceutical/pharmaceutical formulations might be useful concerning better drug performance and efficacy.

Corresponding author: Snehasis Jana, Trivedi Science Research Laboratory Pvt. Ltd., Bhopal, India. Tel: +91-022-25811234; Email: publication@trivedieffect.com

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Introduction

Folic acid (vitamin B\textsubscript{9}) is the name used for a conjugate of 4-aminobenzoic acid and L-glutamic acid \textit{i.e.}, folate \cite{1}. Its biologically active form is tetrahydrofolic acid that acts similar to the cobalamin. Folic acid is important as it helps in the production of deoxyribonucleic acid (DNA) and also plays various vital functions within the body. The folic acid is synthetic form, while its natural form is folate that is found in food. It is generally used in combination with other B vitamins \cite{2}. The important food sources containing folate are cereals, leafy vegetables such as broccoli, spinach, lettuce, etc.; fruits such as melons, bananas, lemons, \textit{etc.;} okra, legumes, asparagus, yeast, organ meat, mushrooms, and orange and tomato juice \cite{3, 4}.

Folic acid is mainly used in the treatment and prevention of folate deficiency and the other associated complications, such as anaemia and malabsorption due to bowel inability \cite{5}. The use of folic acid is also evident in the prevention of stroke and heart disease, due to its property to reduce homocysteine level in the blood, which might create risk for heart disease. The other uses of folic acid involve the treatment of liver disease, ulcerative colitis, alcoholism, and kidney diseases such as dialysis. It is also used by pregnant women to prevent miscarriage and neural tube defects, such as spina bifida. Some researchers also indicated the use of folic acid in the prevention of colon and cervical cancer. Moreover, folic acid is used for the prevention and treatment of Alzheimer’s disease, memory loss, eye disease age-related macular degeneration, age-related hearing loss, aging, sleep problems, osteoporosis, depression, restless leg syndrome, nerve pain, vitiligo, gum infections, muscle pain, and Fragile-X syndrome, which is an inherited disease \cite{6-9}. Folic acid also helps in reducing the side effects of lometrexol and methotrexate drugs and associated problems \cite{10}.

The physicochemical properties of the drug are important regarding its absorption, distribution, metabolism, and excretion (ADME) profile and thus, it is advised to improve the biological activities and efficacy of the drug by altering its physicochemical properties \cite{11}. Consciousness Energy Healing Treatment is one among such approaches that could be used in modifying the physical, structural, and thermal properties of the drugs \cite{12, 13}. The Biofield Energy Healing is widely accepted as an alternative integrative approach that helps in improving the quality of life by correcting the root cause of the diseases \cite{14-16}. A human can harness energy from the universe and can transmit it to any living organism(s) or non-living object(s) around the globe. The object or recipient always receives the energy and responds in a useful way. This process is known as the Trivedi Effect\textsuperscript{®}. Biofield Energy Healing Treatment \cite{17, 18}. The Trivedi Effect\textsuperscript{®}. Consciousness Energy Healing Treatment has shown its effect on the crops and its productivity, microbes, metals, chemicals, ceramics, nutraceuticals, skin health, bioavailability, bone health, cancer \cite{19-37}, etc. This study was also designed to establish the physicochemical and thermal properties of the treated folic acid and to determine the impact of the Trivedi Effect\textsuperscript{®} on the properties of folic acid as compared to the control one with the help of various analytical techniques.

Materials and Methods

Chemicals and Reagents

Folic acid was purchased from Alfa Aesar, USA; whereas the remaining chemicals used in the experiments were of analytical grade purchased from India.

Consciousness Energy Healing Treatment Strategies

The folic acid sample used in the study was first divided into two parts. The first part of the samples was not given the Biofield Energy Treatment and considered as the control sample. Besides, the second part of the sample was received the Trivedi Effect\textsuperscript{®}-Energy of Consciousness Healing Treatment under standard laboratory conditions for 3 minutes and known as the Biofield Energy Treated folic acid sample. This Biofield Energy Treatment was provided remotely through the healer’s unique energy transmission process by a famous Biofield Energy Healer, Gopal Nayak, India, to the test sample. Later on, the control sample was treated with a “sham” healer for comparison purpose. The “sham” healer did not have any knowledge about the Biofield Energy Treatment. Now, the control and Biofield Energy Treated sample were kept in sealed conditions and characterized using modern analytical techniques.
Characterization

Particle Size Analysis (PSA)

The particle size analysis of folic acid samples was carried out on Malvern Mastersizer 2000, from the UK with a detection range between 0.01 µm to 3000 µm using wet method [38, 39]. The % change in particle size (d) at below 10% level (d_{10}), 50% level (d_{50}), 90% level (d_{90}), and D(4,3) was calculated using the following equation 1:

\[
\text{% change in particle size} = \left( \frac{d_{\text{Treated}} - d_{\text{Control}}}{d_{\text{Control}}} \right) \times 100
\]

(1)

Where \( d_{\text{Control}} \) and \( d_{\text{Treated}} \) are the particle size (µm) of the control and the Biofield Energy Treated folic acid samples, respectively.

The % change in surface area (S) was calculated using following equation 2:

\[
\text{% change in surface area} = \left( \frac{S_{\text{Treated}} - S_{\text{Control}}}{S_{\text{Control}}} \right) \times 100
\]

(2)

Where \( S_{\text{Control}} \) and \( S_{\text{Treated}} \) are the surface area of the control and treated folic acid samples, respectively.

Powder X-ray Diffraction (PXRD) Analysis

The PXRD analysis of control and the Biofield Energy Treated folic acid was performed with the help of Rigaku MiniFlex-II Desktop X-ray diffractometer (Japan) [40, 41]. The average size of individual crystallites was calculated from XRD data using the Scherrer’s formula 3:

\[
G = \frac{k \lambda}{\beta \cos \theta}
\]

(3)

Where \( k \) is the equipment constant (0.94), \( G \) is the crystallite size in nm, \( \lambda \) is the radiation wavelength (0.154056 nm for Kα1 emission), \( \beta \) is the full-width at half maximum (FWHM), and \( \theta \) is the Bragg angle [42].

The % change in crystallite size (G) of folic acid was calculated using the following equation 4:

\[
\text{% change in crystallite size} = \left( \frac{G_{\text{Treated}} - G_{\text{Control}}}{G_{\text{Control}}} \right) \times 100
\]

(4)

Where \( G_{\text{Control}} \) and \( G_{\text{Treated}} \) are the crystallite size of the control and treated samples, respectively.

Thermal Gravimetric Analysis (TGA)/ Differential thermogravimetric analysis (DTG)

TGA/DTG thermograms of the control and the Biofield Energy Treated folic acid were obtained with the help of TGA Q50TA instruments. A sample of 4-15 mg was loaded to the platinum crucible at a heating rate of 10ºC/min from 25ºC to 1000ºC with the recent literature [43]. The % change in weight loss (W) was calculated using the following equation 5:

\[
\text{% change in weight loss} = \left( \frac{W_{\text{Treated}} - W_{\text{Control}}}{W_{\text{Control}}} \right) \times 100
\]

(5)

Where \( W_{\text{Control}} \) and \( W_{\text{Treated}} \) are the weight loss of the control and the Biofield Energy Treated folic acid, respectively.

The % change in maximum thermal degradation temperature (\( T_{\text{max}} \)) (M) was calculated using the following equation 6:

\[
\text{% change in } T_{\text{max}} (M) = \left( \frac{M_{\text{Treated}} - M_{\text{Control}}}{M_{\text{Control}}} \right) \times 100
\]

(6)

Where \( M_{\text{Control}} \) and \( M_{\text{Treated}} \) are the \( T_{\text{max}} \) values of the control and treated folic acid, respectively.

Differential Scanning Calorimetry (DSC)

The DSC analysis of folic acid was performed with the help of DSC Q200, TA instruments. A sample of ~1-5 mg was loaded to the aluminium sample pan at a heating rate of 10ºC/min from 30ºC to 350ºC [43]. The % change in melting point (T) was calculated using the following equation 7:

\[
\text{% change in melting point} = \left( \frac{T_{\text{Treated}} - T_{\text{Control}}}{T_{\text{Control}}} \right) \times 100
\]

(7)

Where \( T_{\text{Control}} \) and \( T_{\text{Treated}} \) are the melting point of the control and treated samples, respectively.

The % change in the latent heat of fusion (\( \Delta H \)) was calculated using the following equation 8:

\[
\text{% change in the latent heat of fusion} = \left( \frac{\Delta H_{\text{Treated}} - \Delta H_{\text{Control}}}{\Delta H_{\text{Control}}} \right) \times 100
\]

(8)

Where \( \Delta H_{\text{Control}} \) and \( \Delta H_{\text{Treated}} \) are the latent heat of fusion of the control and treated folic acid, respectively.

Statistical Analysis

The values were represented as Mean ± SEM (standard error of mean) of the independent experiments. For two groups comparison student’s t-
test was used. Statistically significant values were set at the level of \( p \leq 0.05 \).

**Results and Discussion**

**Particle Size Analysis (PSA)**

The particle size analysis corresponding to \( d_{10} \), \( d_{50} \), \( d_{90} \), and \( D(4, 3) \) of the control and the Biofield Energy Treated sample was done and the results are mentioned in Table 1. It revealed that the particle size distribution of the treated folic acid sample was significantly reduced by 7.01\%, 6.53\%, 8.37\%, and 15.99\% at \( d_{10} \), \( d_{50} \), \( d_{90} \), and \( D(4, 3) \), respectively, compared to the control sample.

The decrease in particle size resulted in the increase in specific surface area (SSA) of the treated sample as the SSA was observed as 1.74 m\(^2\)/g in the control sample; while it was increased by 1.72\% and observed as 1.77 m\(^2\)/g in the treated sample. The particle size distribution is important regarding the efficacy and performance of drug within the body in terms of its solubility, dissolution, and bioavailability [11, 44]. The decreased particle size of the treated folic acid that resulted in the increased surface area could be used as an approach to improve the bioavailability and efficacy of drug [45]. Hence, it could be anticipated that the treated folic acid might show better bioavailability profile after the Biofield Energy Treatment when used in the pharmaceutical and nutraceutical formulations in comparison to the untreated sample.

**Powder X-ray Diffraction (PXRD) Analysis**

The PXRD diffractograms of the control and treated folic acid samples are shown in Figure 1. The analysis revealed the crystalline nature of both the samples as shown by the sharp and intense peaks present in their diffractograms. The further analysis was done (Table 2) to determine any changes in the Bragg’s angles, relative peak intensities, and crystallite sizes corresponding to the characteristic peaks of the treated sample as compared to the control sample.

The results revealed that the Bragg’s angles of the peaks present in the diffractogram of the treated sample were slightly altered compared to the control sample; however the major alteration was observed in the peak at 2\( \theta \) equals to 12.29\( ^{\circ} \) in the control sample that was observed at 2\( \theta \) equals to 13.07\( ^{\circ} \) (entry no. 2) in the treated sample’s diffractogram. Besides, the peak intensities of the treated folic acid sample were altered ranging from -36.81\% to 113.41\%; and the crystallite sizes were varied ranging from -15.79\% to 318\%, compared to the control sample. The major alteration was also observed in the average crystallite size as it was observed as 152.68 nm in the treated sample, which was significantly increased by 22.55\% in comparison to the control sample (124.58 nm). The alterations in the crystalline structure, as well as crystal morphology of drugs have been reported previously by using the Biofield Energy Treatment. It is supposed that the Consciousness Energy Healing Treatment might form a new polymorph of the compound by changing the peak intensities and crystallite sizes [46, 47]. Such novel polymorph of folic acid might possess better bioavailability and drug profile compared with the untreated sample.

**Thermal Gravimetric Analysis (TGA)/ Differential Thermogravimetric Analysis (DTG)**

The analysis of the thermal stability of both the samples, i.e., the control and treated samples were done using TGA/DTG technique. The TGA thermograms of both the samples are presented in Figure 2 and the results (Table 3) revealed that the total weight loss of the treated folic acid during thermal degradation was increased by 3.94\% as compared to the control sample. Therefore, the residual amount remaining after the degradation of the Biofield Energy Treated sample was observed to be significantly reduced by 14.99\% in comparison to the control folic acid sample. Hence, it indicated the increased thermal degradation of the treated sample compared to the untreated folic acid sample.

The DTG analysis of both the samples showed four peaks in their respective thermograms (Figure 3). It was observed that the maximum thermal degradation temperatures (\( T_{\text{max}} \)) in the treated sample corresponding to 1\( ^{\text{st}} \), 2\( ^{\text{nd}} \) and 3\( ^{\text{rd}} \) peaks were increased by 1.23\%, 0.68\%, and 2.89\%, respectively compared to the control sample. However, the treated sample showed a slight reduction in the \( T_{\text{max}} \) of the 4\( ^{\text{th}} \) peak by 1.39\% in comparison to the control folic acid sample. Thus, the overall analysis revealed improvement in the thermal degradation temperature of the Biofield Energy Treated
Table 1. Particle size distribution of the control and the Biofield Energy Treated folic acid.

| Parameter          | \(d_{10}\) (µm) | \(d_{50}\) (µm) | \(d_{90}\) (µm) | \(D(4,3)\) (µm) | SSA (m²/g) |
|--------------------|------------------|------------------|------------------|------------------|------------|
| Control            | 1.71             | 4.75             | 17.81            | 8.69             | 1.74       |
| Biofield Treated   | 1.59             | 4.44             | 16.32            | 7.30             | 1.77       |
| Percent change* (%)| -7.01            | -6.53            | -8.37            | -15.99           | 1.72       |

\(d_{10}, d_{50},\) and \(d_{90}\): particle diameter corresponding to 10%, 50%, and 90% of the cumulative distribution, \(D(4,3)\): the average mass-volume diameter, and SSA: the specific surface area. *denotes the percentage change in the Particle size distribution of the Biofield Energy Treated sample with respect to the control sample.

Figure 1. PXRD diffractograms of the control and the Biofield Energy Treated folic acid.
Table 2. PXRD data for the control and the Biofield Energy Treated folic acid.

| Entry No. | Bragg angle (°2θ) | Intensity (cps) | Crystallite size (G, nm) |
|-----------|-------------------|-----------------|--------------------------|
|           | Control | Treated | Control | Treated | % change* | Control | Treated | % change* |
| 1         | 5.46    | 5.49    | 307     | 194     | -36.81    | 99      | 240     | 142.42    |
| 2         | 10.88   | 10.96   | 373     | 559     | 49.87     | 195     | 203     | 4.1       |
| 3         | 12.29   | 13.07   | 261     | 557     | 113.41    | 50      | 209     | 318       |
| 4         | 16.95   | 16.99   | 357     | 491     | 37.54     | 84      | 78.1    | -7.02     |
| 5         | 19.23   | 19.31   | 91      | 118     | 29.67     | 227     | 218     | -3.96     |
| 6         | 20.45   | 20.55   | 188     | 237     | 26.06     | 114     | 96      | -15.79    |
| 7         | 21.68   | 21.76   | 112     | 117     | 4.46      | 181     | 212     | 17.13     |
| 8         | 26.66   | 26.69   | 593     | 870     | 46.71     | 110     | 110     | 0         |
| 9         | 27.75   | 27.77   | 253     | 470     | 85.77     | 199     | 168     | -15.58    |
| 10        | 29.6    | 29.52   | 565     | 436     | -22.83    | 42      | 92      | 119.05    |
| 11        | 31.27   | 31.31   | 109     | 145     | 33.03     | 103     | 120     | 16.5      |
| 12        | 34.34   | 34.46   | 111     | 146     | 31.53     | 91      | 86      | -5.49     |
| 13        |         |         |         |         |           | 124.58 ± 17.57 | 152.68 ± 17.65 | 22.55 |

* denotes the percentage change in the crystallite size of the Biofield Energy Treated sample with respect to the control sample, SEM: standard error of the mean.

Table 3. TGA/DTG data of the control and the Biofield Energy Treated samples of folic acid.

| Sample                  | TGA | DTG {T<sub>max</sub> (°C)} |
|-------------------------|-----|-----------------------------|
|                         | Total weight loss (%) | Residue % | Peak 1 | Peak 2 | Peak 3 | Peak 4 |
| Control                 | 79.19 | 20.81 | 107.71 | 247.93 | 433.78 | 709.61 |
| Biofield Energy Treated | 82.31 | 17.69 | 109.04 | 249.63 | 446.32 | 699.77 |
| % Change*               | 3.94 | -14.99 | 1.23 | 0.68 | 2.89 | -1.39 |

* denotes the percentage change of the Biofield Energy Treated sample with respect to the control sample, T<sub>max</sub> = the temperature at which maximum weight loss takes place in TG or peak temperature in DTG.
Figure 2. TGA thermograms of the control and the Biofield Energy Treated folic acid.
Figure 3. DTG thermograms of the control and the Biofield Energy Treated folic acid.
sample compared with the control sample.

**Differential Scanning Calorimetry (DSC) Analysis**

The DSC analysis was used in this study to determine the thermal behaviour of the samples such as melting, crystallization temperature, *etc.* [48]. According to the scientific literature, the heating of folic acid causes the breaking of the "Glu" moiety first at ~180°C, after which the degradation of pterin and PABA takes place in an overlapping form. Moreover, upon further heating, there was the loss of amide and acid functionalities of the compound at ~195 °C, and heating beyond that degrades the crystalline folic acid in the form of amorphous one above 200°C [49].

There were two peaks in the DSC thermograms of the control and treated sample (Figure 4) and the further analysis related to peak temperature and enthalpy change for both the samples was presented in Table 4. The first peak, i.e., endothermic in nature, was observed in both the thermograms that indicated the melting of the control and treated sample. The peak temperature for the treated sample was observed to be increased by 2.42%, along with 10.33% reduction in the latent heat of fusion ($\Delta H_{\text{fusion}}$) compared to the control sample (Table 4). Besides, the second peak in both the samples is broad and exothermic in nature and might indicate the degradation of the samples after further heating. The analysis showed that the degradation temperature of the treated sample was at 231.11°C, i.e., reduced by 1.60% in comparison to the control sample (234.86°C). Moreover, the latent heat of degradation ($\Delta H_{\text{degradation}}$) corresponding to this peak of the treated folic acid sample was significantly increased by 10.46% in comparison to the control sample (Table 4).

The DSC results denoted that the thermal properties of treated folic acid were altered after the Biofield Energy Treatment that might occur due to some changes in the crystallization structure [50] of the treated folic acid as evident in the PXRD analysis. Hence, it is presumed that the Biofield Energy Treated folic acid might be more stable against heating compared to the untreated sample.

**Conclusions**

The study of Biofield Energy Treated folic acid revealed the significant effect of the Trived Effect®-Consciousness Energy Healing Treatment on its physicochemical and thermal properties. It indicated the significant changes in the particle size distribution of folic acid after the Biofield Energy Treatment. The treated sample showed reduced particle size at $d_{10}$, $d_{50}$, $d_{90}$, and D(4,3) by 7.01%, 6.53%, 8.37%, and 15.99%, respectively compared to the control sample. The decrease in particle size of the Biofield Energy Treated sample causes an increase in the specific surface area by 1.72% compared to the control folic acid sample. The PXRD diffractograms indicated significant alterations in the peak intensities and crystallite sizes ranging from 36.81% to 113.41% and -15.79% to 318%, respectively, compared to the untreated sample. The average crystallite size of the treated folic acid sample was found to be increased by 22.55% after the Biofield Energy Treatment in comparison to the control sample. The TGA data showed the increase in total weight loss of the Biofield Energy Treated sample by 3.94% that resulted in significant reduction in residue weight by 14.99%, compared to the control sample. The treated sample showed that the melting temperature and $\Delta H_{\text{fusion}}$ were altered by 2.42% and -10.33%, respectively; while the degradation temperature and $\Delta H_{\text{degradation}}$ were changed by -1.60% and 10.46%, respectively compared to the untreated folic acid sample. The overall data indicated that the Trivedi Effect®-Consciousness Energy Healing Treated folic acid formed a new polymorph that may improve the solubility, dissolution, absorption, and bioavailability along with thermal stability compared to the untreated sample. Thus, it could be presumed that the Biofield Energy Treated folic acid would be more efficacious in the pharmaceutical/nutraceutical preparations regarding the prevention and treatment of several diseases such as Alzheimer’s disease, memory loss, eye disease, age-related macular degeneration, age-related hearing loss, aging, allergic diseases, sleep problems, osteoporosis, depression, restless leg syndrome, nerve pain, vitiligo, gum infections, muscle pain, Fragile-X syndrome, *etc.*

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Figure 4. DSC thermograms of the control and the Biofield Energy Treated folic acid.

Table 4. Comparison of DSC data between the control and the Biofield Energy Treated folic acid.

| Peak       | Description                | Melting Point (°C) | ΔH (J/g) |
|------------|----------------------------|--------------------|----------|
| Peak 1     | Control sample             | 184.82             | 113.30   |
|            | Biofield Treated sample    | 189.29             | 101.60   |
|            | % Change*                  | 2.42               | -10.33   |
| Peak 2     | Control sample             | 234.86             | 54.85    |
|            | Biofield Treated sample    | 231.11             | 60.59    |
|            | % Change*                  | -1.60              | 10.46    |

ΔH: Latent heat of fusion/decomposition; *denotes the percentage change of the Biofield Energy Treated sample with respect to the control sample.
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Conflict of Interest

Authors declare no conflict of interest.

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