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Competition between shear banding and wall slip in wormlike micelles

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The interplay between shear band (SB) formation and boundary conditions (BC) is investigated in wormlike micellar systems (CPyCl–NaSal) using ultrasonic velocimetry coupled to standard rheology in Couette geometry. Time-resolved velocity profiles are recorded during transient strain-controlled experiments in smooth and sand-blasted geometries. For stick BC standard SB is observed, although depending on the degree of micellar entanglement temporal fluctuations are reported in the highly sheared band. For slip BC wall slip occurs only for shear rates larger than the start of the stress plateau. At low entanglement, SB formation is shifted by a constant $\Delta \dot{\gamma}$, while for more entangled systems SB constantly “nucleate and melt.” Micellar orientation gradients at the walls may account for these original features.

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During the past two decades, shear banding (SB), i.e. the shear-induced coexistence of macroscopic bands with widely different viscosities, has been evidenced in a large range of complex fluids [1]. Sheared dispersions of surfactant wormlike micelles have attracted considerable attention due to their practical use in industry, but also because they challenged the physicists to address a nonequilibrium problem with concepts from thermodynamics [2, 3, 4]. Indeed, rheological measurements show that the flow curve of shear-banding systems, i.e. the measured shear stress $\sigma$ vs. the applied shear rate $\dot{\gamma}$, presents a plateau at a well-defined shear stress $\beta$ over a given range of shear rates $\dot{\gamma}$, very similar to the plateau in pressure as a function of overall concentration of a demixed system. As for equilibrium phase transitions, it has been suggested that the flow can be either metastable or unstable for SB formation, depending on the applied shear rate $\dot{\gamma}$. The formation of two coexisting SB, bearing the local shear rates $\dot{\gamma}_1$ and $\dot{\gamma}_2$ that mark respectively the lower and upper limits of the stress plateau, constitutes a pathway for the relaxation of the excess stress in the initially linear flow. Stress relaxation can, however, also occur through apparent wall slip. Slip phenomena are ubiquitous in polymers [5, 6] and soft glassy materials [7]. Wall slip has also been reported in shear-thinning wormlike micellar systems [8, 9] but its connection with SB has been underexposed. Still, information on the interplay between wall slip and a flow instability like SB are essential for fully understanding the behavior of complex fluids.

In this Letter, wall slip is shown to compete with SB formation by offering an alternative route for stress relaxation. We use tunable boundary conditions (BC) at the walls as an experimental tool to probe the effect of wall slip on the flow behavior of cetylpyridinium chloride/sodium salicylate (CPyCl–NaSal) micellar solutions at 6 and 10 wt. % in 0.5 M NaCl brine at 23°C. We enforce “stick” BC by using a rough sand-blasted Plexiglas Couette cell and partial “slip” BC by using a smooth Plexiglas cell [10]. The competition between SB formation and wall slip after shear rate quenches is addressed through simultaneous rheological [11] and time-resolved velocity profiles measurements. For the latter, we use ultrasonic speckle velocimetry (USV) [12] since the sand-blasted cell is not transparent and optical techniques as in [13, 14, 15] would be too difficult to implement. We show that with slip BC, wall slip occurs only for shear rates larger than the start of the stress plateau for both the concentrations under study. The extent to which SB formation is frustrated by wall slip strongly depends, however, on the degree of micellar entanglement. Very large temporal fluctuations are reported in the more concentrated sample for both BC.

The flow curve of 6 wt. % CPyCl–NaSal shown in Fig. 1(a) for stick and slip BC reveals a stress plateau at $\sigma^* \approx 75$ Pa that extends from $\dot{\gamma}_1 \approx 4.5$ to $\dot{\gamma}_2 \approx 22$ s$^{-1}$, with a slight tilt due to the curvature of the Couette cell [16]. Interestingly the flow curve for slip BC does not show such a sharp bend at $\dot{\gamma}_1$ as with stick BC. Fig. 1(b) presents the stress responses for quenches from $\dot{\gamma}_{\text{init}}$ located in the low shear regime to $\dot{\gamma}_{\text{appl}} = 8$ s$^{-1}$ located in the beginning of the stress plateau. Conform to earlier experiments [2, 3], the stress shows a slow decay after an initial overshoot and a few oscillations. As in Ref. [4], we define the amplitude of this slow relaxation as the excess stress $\Delta \sigma = \sigma^\infty - \sigma^\infty$, where $\sigma^\infty$ is the “mechanical” stress at the end of the oscillations and $\sigma^\infty$ is the steady-state shear stress. Although the initial overshoot is more pronounced for stick BC, the stress responses for $t \gtrsim 10$ s are very similar for both BC. Yet, depending on the BC, velocity profiles display radically different behaviors that persist in the steady state. As seen from Fig. 1(c) and (d), linear profiles are recorded just before and after the shear rate quench for both BC. For
the velocity profile at ε stick BC, a high SB develops within a few seconds at ε = δ/e ≃ 0.5, where δ is the width of the SB and e the gap width. The interface then migrates towards its final position in agreement with previous observations [17]. For slip BC, however, SB do not fully develop (see the velocity profile at t = 3.3 s in Fig. 2(d)) and the system rather slips to reach a steady state characterized by a homogeneous shear flow with substantial wall slip (about 40%) at the inner cylinder, as in a recent report on DNA dispersions [17].

Figure 2 provides the analysis of the time-resolved velocity measurements after quenches to γ_{app} = 8 s^{-1}, as in Fig. 1(c) and (d). Each velocity profile was analyzed to extract the true shear rate γ_{true}(t), the proportion of highly sheared material ε(t), and the local shear rates γ_{±}(t) in each SB [17]. As noted above our results for stick BC are consistent with previous data where no significant wall slip was reported [17]. They also reveal two important new features: (i) the presence of noticeable fluctuations in both γ_{true}(t) and γ_{±}(t) while ε(t) and γ_{±}(t) remain roughly constant for t ≥ 30 s and (ii) the fact that the position of the SB settles with the same dynamics as the shear stress. For slip BC, Fig. 2(a) shows that (i) the imposed shear rate γ_{app} cannot be sustained although initially γ_{true} ≃ γ_{app} and (ii) wall slip sets in immediately and has the same time constant as the stress relaxation since γ_{true}(t) and σ(t) follow the same decay. Figs. 2(b) and (c) reveal that shear banding is observed during the build up of wall slip. A high SB is formed with ε ≃ 0.2 and γ_{±} ≃ 10 s^{-1}, a value close to the intrinsic shear rate for stick BC. However, for slip BC, γ_{±} rapidly drops and approaches γ_{0}, which leads to the loss of banding structure (hence the lack of ε(t) and γ_{±} data for t ≥ 40 s) and to linear profiles with γ_{true} ≃ 5 s^{-1} in the steady state. We conclude that at γ_{app} = 8 s^{-1} the excess stress relaxes fully due to wall slip for slip BC, while for stick BC it relaxes by SB formation. Both processes have the same time constants since the decay of γ_{true}(t) for slip BC is the same as the settling of the SB through ε(t) for stick BC. As a consequence the stress relaxations for stick and slip BC are also similar (see the dashed line in Fig. 2(b)).

Quenches were repeated as described above for final shear rates γ_{app} covering almost the whole stress plateau [20]. Fig. 3 presents the steady state values of the true shear rate γ_{true} and the proportion of the high SB ε_{∞}, as well as the amplitude of the relaxation of ε(t) (noted ∆ε and defined in Fig. 3(b)) and that of the stress relaxation ∆σ. As shown by the solid line γ_{true} = γ_{app} in Fig. 3(a), stick BC apply for the sand-blasted cell. Moreover the linear behavior of ε_{∞} vs. γ_{app} is consistent with the “lever rule”: ε_{∞} = (γ_{app} - γ_{1})/(γ_{2} - γ_{1}) (see solid line in Fig. 3(b)) with γ_{1} = 3.4 ± 0.2 s^{-1} and γ_{2} = 22.4 ± 0.5 s^{-1} in satisfactory agreement with both the flow curve and the steady state values of the local shear rates γ_{1} = 4.3 ± 0.3 s^{-1} and γ_{2} = 22 ± 1 s^{-1} measured from the velocity profiles [19]. These observations not only confirm previous results in the absence...
FIG. 3: Steady state values of (a) the true shear rate $\dot{\gamma}_{\text{true}}$ and (b) the proportion of the high SB $\varepsilon$, and amplitudes of the relaxations of (c) the proportion of the high SB $\Delta\varepsilon$ and (d) the excess shear stress $\Delta\sigma$ as a function of the imposed shear rate $\dot{\gamma}_{\text{appl}}$. Solid (open) symbols refer to stick (slip) BC. All data are for 6 wt. % CPyCl–NaSal except for the triangles in (a) which are for 10 wt. % CPyCl–NaSal with slip BC. In (a) the solid line is $\dot{\gamma}_{\text{true}} = \dot{\gamma}_{\text{appl}}$, while the dashed line is $\dot{\gamma}_{\text{true}} = \dot{\gamma}_{\text{appl}} - 3.4\,\text{s}^{-1}$ and the dotted lines indicate $\dot{\gamma}_{\text{true}}$ for 6 and 10 wt. % CPyCl–NaSal. The dotted lines in (b) show the resolution limit for SB detection [19]. The lines in (c) and (d) are to guide the eye.

FIG. 4: Responses to shear rate quenches from $\dot{\gamma}_{\text{init}} = 0.8\,\text{s}^{-1}$ to $\dot{\gamma}_{\text{appl}} = 8\,\text{s}^{-1}$ for stick (black) and slip BC (red) in 10 wt. % CPyCl–NaSal. (a) True shear rate $\dot{\gamma}_{\text{true}}$. The green line is the response to a quench to $\dot{\gamma}_{\text{appl}} = 5\,\text{s}^{-1}$ for slip BC. (b) Proportion of the high SB $\varepsilon(t)$. (c) Shear rate in the high SB $\dot{\gamma}_+(t)$ (solid) and low SB $\dot{\gamma}_-(t)$ (dashed, only stick BC). The vertical dashed lines indicate the times where SB nucleate and melt (see text). The gaps in the slip data in (b) and (c) are due to the lack of SB.

It is interesting to see how the balance between SB and wall slip changes for a more entangled system, e.g. a 10 wt. % CPyCl–NaSal sample, as studied by López-González et al. [11]. The response to shear rate quenches for this system, where $\sigma^\ast \approx 158\,\text{Pa}$, $\dot{\gamma}_1 \approx 1.7\,\text{s}^{-1}$ and $\dot{\gamma}_2 \approx 20\,\text{s}^{-1}$, is plotted in Fig. 4. Even in the sandblasted cell where stick BC are supposed to be valid, the true shear rate never coincides with $\dot{\gamma}_{\text{appl}}$. Moreover considerable fluctuations in $\dot{\gamma}_1$ are observed, while $\dot{\gamma}_2$ remains much smoother. For slip BC and $\dot{\gamma}_{\text{appl}} = 5\,\text{s}^{-1}$, we observe that the sample slips to the shear rate $\dot{\gamma}_1$ at the start of the stress plateau where there is no excess stress (see the green line in Fig. 4(a)). A quench to a higher shear rate of $\dot{\gamma}_{\text{appl}} = 8\,\text{s}^{-1}$ reveals that $\dot{\gamma}_{\text{true}}$ jumps from $\dot{\gamma}_1$ to significantly higher values over short time windows indicated by vertical dashed lines in Fig. 4. Velocity profiles also show that, when $\dot{\gamma}_{\text{true}} > \dot{\gamma}_1$, a small but detectable high SB forms with $\varepsilon \gtrsim 0.1$ and $\dot{\gamma}_+ \approx 15\text{-}20\,\text{s}^{-1}$. In other words the high-shear state is formed over short periods of time and is unstable over longer times, which is reminiscent of “nucleation and melt” events typical of metastability. Since $\dot{\gamma}_{\text{true}} \approx \dot{\gamma}_1$ up to $\dot{\gamma}_{\text{appl}} \approx 12\,\text{s}^{-1}$ (see
open triangles Fig. 3(a)), wall slip dominates SB formation over the full accessible part of the stress plateau.

In summary, we established that under slip BC wall slip is observed only for shear rates larger than the start of the stress plateau, i.e. \( \dot{\gamma} > \dot{\gamma}_1 \). For the 6 wt. % sample, SB formation is suppressed by wall slip over the first \( \Delta \dot{\gamma} \approx 3.4 \) s\(^{-1}\) into the stress plateau, while for higher shear rates stable SB are observed together with partial slip. For 10 wt. % CPyCl-NaSal, “nucleation and melt” is observed over the full accessible part of the stress plateau.

This has the important implication that for both concentrations wall slip acts as to stabilize the bulk flow. By combining presently available theories we may interpret our results along the following line of argumentation. Strong gradients in the shear rate can build up at the wall, assuming that wormlike micelles preferentially align with the smooth walls, i.e. that gradients in orientation are intrinsically present at the wall, see Ref. [22]. Orientation gradients grow when the system is quenched into the plateau region, see Refs. [21]. The stress that is stored in the system after the quench needs to diffuse in order for the system to relax, see Ref. [22]. These latter two processes should be independent of whether the gradients are present in the bulk or at the wall. Combining these arguments one can explain the observation that no apparent wall slip is observed below the stress plateau because in this region the flow is stable and gradients at the walls or in bulk do not grow. It also follows that the time constants of \( \dot{\gamma}_{\text{true}} \) for slip BC and \( \varepsilon \) for stick BC are comparable in Fig. 3, resulting in similar stress decays (see Fig. 3a), since the same stress diffusion is needed in both cases. Once the SB have settled \( \dot{\gamma} = \dot{\gamma}_1 \) holds for the low SB both for stick and slip BC. Fluctuations between both conditions can now easily occur since no stress diffusion is needed. This may account for the difference between the stable low SB and the fluctuating high SB (see Fig. 3c), and for the fast formation of the nucleating bands in the more concentrated sample.

To conclude, BC appear to be a crucial control parameter that accounts for some of the fluctuations reported earlier on similar systems [1] [2]. The interplay between wall slip and SB formation may have major implications for tuning the flow behavior of complex fluids showing flow instabilities. A full understanding of our experiments still requires a proper combination of the above mentioned theories. The competition between local stress relaxation at the wall via slip and bulk relaxation through SB formation depends on the details of the system, such as the surface treatment and the degree of entanglement in the bulk. A possible microscopic input in the theory could be to mimic surface roughness, i.e. stick BC, by randomizing the alignment of the wormlike micelles at the walls. This is missing from the theoretical work so far. Experiments on a less coarse grained level as was achieved here with USV are also needed to identify micellar orientations at the wall.

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