Zinc Oxide Nanoparticles and Nanorods as Antimicrobial Agents: Particle Size Influence

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Abstract
The aim of this study is to find out about the antibacterial activity of a variety of ZnO nanostructures, along with their derivative nanoparticles (ZnO NPs) and nanorods (ZnO NRs) against numerous clinic strains of Gram-negative Escherichia coli (E. coli), Salmonella typhi (S. typhi), Pseudomonas aeruginosa (P. aeruginosa) as well as Gram-positive Streptococcus pneumonia (Strep. pneumonia) bacteria. ZnO NPs and ZnO NRs were efficiently synthesized by using sol-gel and hydrothermal strategies, respectively. Various properties, consisting of the morphology, structure and optical, had been described by using field emission scanning electron microscopy (FESEM), X-ray diffraction (XRD) as well as ultraviolet-visible (UV-Vis) spectroscopy analysis. This check was performed in Mueller-Hinton agar. The impact of different particle sizes and morphologies of ZnO nanomaterials on the growth of bacteria was measured. In the antibacterial assay, each type of ZnO nanostructure showed effective inhibition. The findings showed that ZnO NRs exhibited more efficient antibacterial activity than that of ZnO NPs agents. This was once the case for both Gram-positive and Gram-negative bacteria. Hence, the study indicated that the antibacterial of ZnO NRs against S. pneumonia was similar to those for S. typhi. However, depending on the particle size effect of ZnO nanostructure, it was found that ZnO NPs showed much less antibacterial activity towards S. typhi than ZnO NRs did.

Keywords: ZnO nanoparticles; ZnO nanorods; Particle size; Antibacterial activity

Introduction
As well known in scientific fields, zinc oxide (ZnO) is a metal oxide with a wide variety of applications in various fields like optics, piezoelectricity, sensing of gas and magnetics. One of the key features of zinc oxide nanostructures is its high catalytic effectiveness, powerful adsorption ability as well as its widespread utilization in sunscreen manufacture [1]. The nanoparticles of zinc oxide, i.e. ZnO NPs, have been globally synthesized due to these convenient properties, justifying their utilization in many diverse fields, including nano diagnostics [2, 3], nanomedicine [4, 5], antimicrobials [6], and immune toxicity [7]. ZnO NPs have gotten noteworthy attention because of their additional usual ultraviolet (UV) filtering characteristic, antibacterial and high catalytic activities, photochemical antifungal. Moreover, it is a provocative material for applications in gadgets, photonics, acoustics, photovoltaic and photodetectors [8]. Most significant properties of ZnO have been stated in the work by Abd Elhady [9], where is was inferred

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that ZnO nanorods (ZnO NRs) had UV protection phenomena and antibacterial activities. A various, low-cost, easy, and considerable of techniques have been utilized to prepare ZnO in distinct morphologies and sizes for diverse applications, such as oxidation of metallic zinc powder [10], sol-gel method [11], pulsed laser deposition [12], hydrothermal synthesis [13], chemical vapor deposition [14], electrodeposition [15], and a sputtering method [16].

The antimicrobial activity of zinc oxide nanoparticles has been studied by various researchers [17-19]. But we used two ZnO nanostructures (nanoparticles and nanorods) in the present study to investigate the impact of particle size and morphology on antimicrobial performance. Therefore, ZnO has been put under the spotlight lately due to its easiness to prepare and its antimicrobial and antifungal processes even with lower concentrations. These features make it feasible for various thin coating applications as well [20]. As claimed via the U.S. Food and Drug Administration (FDA) (21CFR182.8991), ZnO is dubbed “generally documented as safe” (GRAS). Further, they have been shown to possess a diverse set of antibacterial processes against Gram-positive and -negative bacteria, comprising predominant foodborne pathogens such as Salmonella, Escherichia coli O157: H7, and Staphylococcus aureus [21, 22]. Utilization of antibiotics poses many facet effects, some of which consist of a rise in bacterial resistance [23, 24]. Due to the aforementioned details, the current study sought to check the inhibitory impact of ZnO NPs and ZnO NRs towards bacteria.

**Experimental**

**Preparation of ZnO nanoparticles and ZnO nanorod arrays**

Sol-gel approach was utilized to set up zinc oxide nanoparticles through mixing 0.1 M of ethanolic solution of zinc acetate alongside with 0.1 M of diethanolamine. In order to nurture a homogeneous and secure colloid solution, the mixture was stirred at 60 °C for 30 min. Afterward, it was left to ensure getting aging overnight. Furthermore, hydrothermal manner was used to prepare zinc oxide nanorods. It was the subsequent method to put together zinc oxide nanoparticles by using mixing an aqueous solution of 0.04 M zinc nitrate Zn(NO₃)₂ and 0.04 M hexamethylenetetramine (HMTA). The hydrothermal method was executed at the temperature of 90 °C with the employment of a digital oven and constant increase time of 4 h.

**Antibacterial**

Antibacterial activity of zinc oxide nanoparticles and nanorods was examined against hospital strain of Gram-negative E. coli, S. typhi, P. aeruginosa as well as Gram-positive Strept. pneumonia bacteria. These bacteria were received from the Diwaniyah Teaching Hospital and ensured to be diagnostic in the University of Al-Qadisiyah/College of Biotechnology/Iraq. Strains were maintained on nutrient agar for further study.

**Agar well diffusion assay**

The Agar well diffusion approach was utilized to filter the antibacterial endeavor of the synthesized ZnO NPs and ZnO NRs. Wells that had been constructed using a sterile cork borer (6 mm) bored into the Mueller-Hinton Agar (Himedia). Bacterial suspensions were organized with a turbidity of 0.5 McFarland standards (cell density =1 × 10⁸ CFU/Ml) and 100 µL from suspension were swabbed onto the petri dish. The wells were filled with 100 µL ZnO NPs and ZnO NRs from a range of concentrations of 25, 50, 75, 100% by means of micropipette and plates. Next, they were incubated at 37 °C for 24 h. The antibacterial activity was estimated by calculating the inhibition zone of the check microorganisms. All the exams were performed in triplicates underneath sterile condition. Gentamicin was once utilized as an advantageous control.

**Characterizations**

The crystallite structures of zinc oxide nanoparticles and nanorods were investigated using X-ray diffractometer (XRD) (Panalytical X’Pert Pro) by using CuKα radiation (λ = 1.54 Å) at 40 kV and 40 mA. Moreover, the samples’ morphology was examined through field emission scanning electron microscopy (FESEM) analysis (JEOL JSM-7600F). Next, the absorbance spectra of the equipped samples were tested by Perkin Elmer Lambda 20 UV-Visible spectrophotometer.

**Results and Discussion**

**XRD analysis**

Fig. 1 illustrates the XRD results for ZnO nanoparticles and nanorods which were heat-treated using indium tin oxide (ITO) glass substrates. Each sample’s peaks may be indexed as pure zinc oxide wurtzite structure [JCPDS: 00-003-0888]. In addition
to these peaks, only ITO peaks [JCPDS: 01-089-4598] that belongs to substrates were observed. Starting of the X-ray line broadening of zinc oxide nanoparticles contrasted with ZnO nanorods, the crystallization sizes were obtained using $D = \frac{0.94 \lambda}{\beta \cos \theta}$: where $\lambda$ is the X-ray wavelength, $\theta$ is the Bragg angle, and $\beta$ is the full width at half maximum (FWHM) of the diffraction peak which was 19.1 nm, 22.2 nm, respectively. The rise of (002) peak for zinc-oxide samples exhibiting anisotropic behavior could be observed in Fig. 1(a) and (b) which confirmed that the suitable orientation of both samples was in the c-axis direction.

**FESEM images**

Multiple morphologies of zinc oxide nanoparticles and nanorods were made quite apparent, as shown in Fig. 2 displaying FESEM images. Moreover, the mean samples’ particle size was located to be 20±5 nm, while for ZnO NRs, the mean length was observed to be 700±50 nm and the diameter to be 60±5 nm. As expected, surface area of the ZnO NRs was found to be much greater than that of ZnO NPs, which is apparent in Fig. 2(b). Additionally, the particle size values of both ZnO morphologies could be clearly observed.

**Ultraviolet–visible spectrum**

The plot of absorbance spectra for both zinc oxide nanoparticles and nanorods are displayed in Fig. 3. It could be noted that the absorbance edge shifted a little towards longer wavelengths. Thus, it could be inferred that the absorbance of materials was influenced by particle size. The absorbance spectra showed absorption peaks of ZnO nanoparticles and nanorods at wavelengths of 387 nm and 393 nm, respectively, which could be attributed to the fundamental band-gap of ZnO (as shown in the inset Fig. 3) owing to the transition of electrons from the valence band (VB) to the conduction band (C-B).

**Antimicrobial activity**

The antibacterial impact of ZnO (NPs and NRs) was gauged via obtaining the zone of inhibition (displayed in Table 1) versus the studied pathogenic strains. Findings showed that zinc oxide nanoparticles and nanorods exhibited effective antibacterial activity for all strains tested. In general, the findings illustrated that the inhibitory effect of zinc oxide nanoparticles and nanorods rose proportionally with higher concentrations. The bacteria tested exhibited
higher sensitivity towards the synthesized zinc oxide nanoparticles and nanorods (Fig. 4 and 5). All the tested bacteria were found to be relatively more sensitive to zinc oxide nanoparticles and nanorods when compared to antibiotics. Similar findings were shown by Fair and Tor [25], where it was shown that the synthesized zinc oxide nanoparticles exhibited improved antimicrobial activities than synthetic drugs did for Gram-positive and -negative bacteria. The findings have confirmed that the small size of nanoparticles is supporting the ease of entering microbial cell membrane and thereby allowing inhibition mechanisms to activate within the cell [26]. It may also be explained by the involvement of reactive oxygen species (ROS) production [27, 28], and accumulation of nanoparticles within the cytoplasm or outer membranes [29, 30], disruption of the cell membrane and oxidative stress [31]. Findings illustrate that Gram-positive bacteria (Strept. pneumoniae) exhibited higher sensitivity to zinc oxide nanoparticles and nanorods than Gram-negative bacteria. Another explanation may additionally be due to the differing nature of surface membranes for the range of bacteria current [32].

**Table 1** Antimicrobial activity of ZnO NRs and ZnO NPs against different bacteria. Diameter of inhibition zone is mm. G: Gentamycin

| Sample          | Inhibition zone diameter (mm) |
|-----------------|-------------------------------|
|                 | E. coli | S. typhi | P. aeruginosa | Strept. pneumonia |
| 25% ZnO NPs    | 10      | 0.0      | 11            | 29               |
| 50% ZnO NPs    | 12      | 0.0      | 16            | 29               |
| 75% ZnO NPs    | 14      | 14       | 18            | 30               |
| 100% ZnO NPs   | 23      | 15       | 17            | 30               |
| 25% ZnO NRs    | 13      | 12       | 11            | 12               |
| 50% ZnO NRs    | 15      | 30       | 16            | 29               |
| 75% ZnO NRs    | 18      | 30       | 16.5          | 30               |
| 100% ZnO NRs   | 30      | 40       | 20            | 40               |
| G               | 20      | 17       | 17            | 22               |

**Fig. 4** Antibacterial activity of ZnO NPs against (a) E. coli, (b) Strept. Pneumonia, (c) S. typhi, and (d) P. aeruginosa, at different concentrations. G: Antibiotic (Gentamycin); DIW: Deionize water.

**Fig. 5** Antibacterial activity of ZnO NRs against (a) E. coli, (b) S. typhi, (c) P. aeruginosa, and (d) Strept. Pneumonia at different concentrations. G: Antibiotic (Gentamycin); DIW: Deionize water.
The previous work results exhibited that the antibacterial activity of ZnO NPs increased when sizes of the particle were small with high concentrations based on the time effect. Additionally, ZnO nanorods were observed to have been further effective than ZnO nanoparticles. One clarification for this finding might be because of antibacterial impact contingent upon some parameters, for example, particle size, morphology, surface functionality, and fabrication, that has resulted from the large of the surface area of ZnO NRs compared with ZnO NPs. Because of the increasing of touching the surface area that is a significant condition for the effects of nanoparticles shapes [33, 34], consequently, the functional activity of nanoparticles has been strongly influenced by their size. Based on this, the antimicrobial activity of ZnO on bacteria has been enhanced with a diminution of particle size [35, 36]. This used to be interpreted due to the fact of an excess in the surface area to volume, which resulted in the increment reactivity of ZnO surface in nanometer-size. Previous studies [37-39] reported that aqueous suspensions of small nanoparticles of ZnO produced improved levels of reactive oxygen species, particularly hydroxyl radicals with the accumulation of nanoparticles.

It may additionally be concluded from the consequence that ZnO NPs and ZnO NRs are well contenders for performing as antibacterial agents for each Gram-positive and Gram-negative bacterium. Based on that, antimicrobial casings are of significant attention for safeguard of the surfaces, because of the existence of microorganisms on superfcies in the environment can diffuse diseases [40].

Conclusions

The multitude of antibacterial activity for zinc oxide nanomaterials possessing various morphology was highlighted in the current study. The antibacterial activity of ZnO nanorods was found to be greater than their nanoparticle counterparts. Variations in antibacterial activity against _S. typhi_ was found to be commensurate to those against _E. coli_. Nevertheless, when considering the particle size impact on impedance microbiology, Letch-antibacterial activity was found to be much less than its worth and closer to _S. typhi_. Hence, the occurrence of antibacterial activity may need to be defined via the generation of H_2O_2 from the surface of ZnO.

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