The effect of biological particles and their ageing processes on 
aerosol radiative properties: 
Model sensitivity studies

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Abstract. Biological aerosol particles (BAPs) such as bacteria, viruses, fungi and pollen, represent a small fraction of the total aerosol burden. However due to their unique properties, they have been suggested to be important in for radiative forcing by the aerosol direct and indirect effects. By means of process model studies, we compare the sensitivity of these radiative effects to various physicochemical BAP properties (e.g. number concentration, diameter, hygroscopicity, surface tension, contact angle between ice and particles). Exceeding previous sensitivity studies, we explore not only the variability of these properties among different BAP types, but also the extent to which chemical (e.g. nitrification), physical (e.g. fragmentation) and biological (e.g. bacteria cell generation) ageing processes of BAPs can modify these properties. Our model results lead to a ranking of the various properties for the radiative effects: (i) Given that BAPs contribute ~0.1% to total cloud condensation nuclei (CCN) number concentration, their effect on total CCN is likely small. (ii) BAPs number fraction of large particles (diameter > ~0.5 µm) is much higher, resulting in a relatively more important effect on direct radiative forcing. (iii) In mixed-phase clouds at T > -10 °C, BAPs can contribute ~100% to ice nuclei (IN), which makes their role as IN the most important. Our study highlights the need of implementing ageing processes of different BAPs into models as BAP size, CCN and IN activity and optical properties may be sufficiently altered to affect BAP’s residence time and survival in the atmosphere. In particular, we suggest the potential role of biological processes, that are currently not included in aerosol models due to the sparsity of comprehensive data, could affect physicochemical BAP properties.
1 Introduction

Biological aerosol particles (BAPs) contribute a small fraction (50 Tg yr\(^{-1}\), with an upper limit of 1000 Tg yr\(^{-1}\)) to the total natural global aerosol emissions of \(~2900\text{-}13000\text{ Tg yr}^{-1}\) (Stocker et al., 2013). They consist of bacteria, proteins, viruses, fungi, pollen and other biologically-derived materials with potentially infectious, allergenic, or toxic properties (Fröhlich-Nowoisky et al., 2016). They have attracted great interest in the atmospheric science and public health community as they might affect the climate and be responsible for spreading diseases (Asadi et al., 2020; Behzad et al., 2018).

Their mass (Graham et al., 2003; Heald and Spracklen, 2009), number concentrations (Huffman et al., 2013; Matthias-Maser et al., 1999), and fractions (Jaenicke, 2005) can greatly vary depending on the location (Schumacher et al., 2013; Shen et al., 2019; Wei et al., 2016; Yu et al., 2016), time of day (Kang et al., 2012) and other conditions (Graham et al., 2003; Jiaxian et al., 2019; Wu et al., 2016). For example, in the Amazonian rainforest, BAPs contribute \(~20\%\) to the mass of submicron organic aerosol (Schneider et al., 2011). In an urban area in Mainz in central Europe, the number fraction was 5-50\% for particles with diameter (D) > 0.4 \(\mu\text{m}\) (Jaenicke, 2005). Above the ocean, 1\% of particles with 0.2 \(\mu\text{m} < D < 0.7 \mu\text{m}\) contain biological material (Pósfai et al., 1998). Temporal variability of BAPs was observed exhibiting peaks in the morning, during and after rain (Huffman et al., 2013; Zhang et al., 2019). To the total global BAP emissions, bacteria contribute 0.4-1.8 Tg yr\(^{-1}\), which is less than 25-31 Tg yr\(^{-1}\) by fungal spores (Heald and Spracklen, 2009; Hoose et al., 2010) and 47 Tg yr\(^{-1}\) by pollen (Burrows et al., 2009a, 2009b). Although the mass fraction of bacteria is small, their number concentration (\(~0.001\text{-}1 \text{ cm}^{-3}\)) (Lighthart and Shaffer, 1995; Tong and Lighthart, 2000) is larger than that of fungal spores (\(~0.001\text{-}0.01 \text{ cm}^{-3}\)) and pollen (\(~0.001 \text{ cm}^{-3}\)) (Huffman et al., 2010). The concentration of viruses can reach up to \(~0.1 \text{ cm}^{-3}\) in indoor air (Prussin et al., 2015) and decreases to \(~0.01 \text{ cm}^{-3}\) outdoors (Després et al., 2012; Weesendorp et al., 2008). The comparably small size of viruses and bacteria (\(D_{\text{viruses}} \sim 0.1 \mu\text{m}\), \(D_{\text{bacteria}} \sim 1 \mu\text{m}\), \(D_{\text{pollen}} \sim 10 \mu\text{m}\)) enables relatively long residence times of several days in the atmosphere (Burrows et al., 2009a; Verreault et al., 2008).

In numerous recent review articles, it has been suggested that BAPs can affect radiative forcing in multiple ways (Figure 1) (Coluzzo et al., 2017; Després et al., 2012; Haddrell and Thomas, 2017; Hu et al., 2018; Šantl-Temkiv et al., 2020; Smets et al., 2016): BAPs might directly interact with radiation by scattering or absorbing light (Figure 1a). While their aerosol direct effect is likely globally small due to low BAP number concentration (Löndahl et al., 2014), it may be of greater interest locally and for specific wavelength ranges due to the large size of BAPs (Myhre et al., 2013). The optical properties of BAPs (Arakawa et al., 2003; Hu et al., 2019; Thrush et al., 2010) resemble those of other organic particles as BAPs are largely composed of proteins and other macromolecules. Accordingly, BAPs’ optical properties can be ascribed to specific organic functional entities such as amino groups or aromatic structures (Hill et al., 2015; Hu et al., 2019).
At subsaturated relative humidity (RH) conditions, the hygroscopicity ($\kappa_{\text{BAP}}$) determines their ability to take up water (Petters and Kreidenweis, 2007) and thus their equilibrium size, which affects their direct radiative properties. Their hygroscopicity shows a large range ($0.03 \leq \kappa_{\text{BAP}} \leq 0.25$), which is explained by variation of surface composition due to different types of BAPs and/or ageing processes (Bauer et al., 2003; Haddrell and Thomas, 2017; Šantl-Temkiv et al., 2020; Sun and Ariya, 2006).

**Figure 1.** Schematic of the influence of BAP properties and ageing processes on direct and indirect radiative effects. (a) The direct radiative forcing is influenced by BAP concentration ($N_{\text{BAP}}$), diameter ($D_{\text{BAP}}$), refractive index ($n_{\text{BAP}} = n + ik$), surface tension ($\sigma_{\text{BAP}}$), and hygroscopicity ($\kappa_{\text{BAP}}$) affect scattering/absorption of aerosol populations at RH < 100%. (b) $N_{\text{BAP}}$, $D_{\text{BAP}}$, surface tension of aqueous particles ($\sigma_{\text{BAP}}$), and hygroscopicity ($\kappa_{\text{BAP}}$) affect CCN activity and properties of warm clouds. (c) $N_{\text{BAP}}$, $D_{\text{BAP}}$, and contact angle of ice germ on the substrate ($\theta_{\text{BAP}}$) affect the evolution of mixed-phase clouds.

Convective and precipitating clouds lead to efficient particle redistribution by vertical transport and removal of particles by wet deposition. Therefore, cloud-related physicochemical properties need to be constrained to determine the distribution and residence time of BAPs in the atmosphere. Since BAPs often have supermicron sizes, they may act as ‘giant CCN’ and thus induce early precipitation (Barahona et al., 2010; DeLeon-Rodriguez et al., 2013; Feingold et al., 1999). Based on a global model study of BAPs, it was concluded that CCN-relevant properties need to be refined in order to further probe their role in the climate system (Konstantinidis, 2014). Several experimental studies have explored the CCN properties of BAPs and determined their hygroscopicity ($\kappa$) (Ariya et al., 2009; Sun and Ariya, 2006). The role of biosurfactant production by bacteria and fungi has been also discussed in the context of their CCN activity since a lower surface tension ($\sigma_{\text{BAP}}$) enhances water uptake (Renard et al., 2016). In addition, biosurfactant molecules that are produced by bacteria and fungi, while they reside on leaves or other surfaces, might attach to other particles, thus, increasing their CCN ability as well.
In addition to acting as CCN, some species of plant pathogen bacteria and fungi can nucleate ice at $T > -10^\circ C$ (Hoose and Möhler, 2012; Morris et al., 2004, 2008; Pouzet et al., 2017), which makes them unique in terms of ice nucleation to affect the evolution of mixed-phase clouds at these temperatures (Figure 1c). Above vegetated forests (Tobo et al., 2013) and near the surface of the Southern Ocean (Burrows et al., 2013), BAPs have been shown to contribute significantly to the total abundance of IN: In a high altitude mountain region of the United States, ambient measurements suggest that 16-76% of IN at -30 °C consist of primary biological material (Pratt et al., 2009); a similar proportion (33%) was reported at -31 to -34 °C in the Amazon basin (Prenni et al., 2009).

The radiative impacts of BAPs, influenced by the physicochemical properties ($N_{\text{BAP}}$, $D_{\text{BAP}}$, $\kappa_{\text{BAP}}$, $\sigma_{\text{BAP}}$, $m_{\text{BAP}}$, $\theta_{\text{BAP}}$), summarized in Figure 1, can largely differ on spatial and temporal scales, leading to different conclusions regarding the climatic impacts of BAPs (Burrows et al., 2009a, 2009b; Hoose et al., 2010; Junge and Swanson, 2008; Konstantinidis, 2014; Sahyoun et al., 2017; Sesartic et al., 2012). These properties are even more variable than represented in current models as BAPs undergo chemical, physical and biological ageing processes (Coluzza et al., 2017; Deguillaume et al., 2008; Pöschl, 2005; Vaïtilingom et al., 2010).

- **Physical** transformations include agglomeration/fragmentation of cells (Coluzza et al., 2017; Lighthart, 1997; Zhang et al., 2019), coating with organic or inorganic components (Pöschl and Shiraiwa, 2015; Joly et al., 2015), or with solid ice or liquid water (Joly et al., 2013). These processes might alter various physicochemical properties listed in Figure 1.

- **Chemical** transformations include oxidation (Jayaraman et al., 2008; Vaïtilingom et al., 2010), nitration (Franze et al., 2005), oligomerization (Tolocka et al., 2004), degradation of macromolecules (Estillore et al., 2016), and changes of the protein conformations due to exposure to different pH values (Kristinsson and Hultin, 2004). These reactions lead to the modification of the protein structures and other macromolecules and thus affect BAP optical properties (Myhre et al., 2013), CCN activity (Sun and Ariya, 2006), and IN ability (Attard et al., 2012; Kunert et al., 2019).

- **Biological** processes might be initiated by living microorganisms in BAPs, unlike in other aerosol particles in the atmosphere (Amato et al., 2017; Delort et al., 2017; Joly et al., 2015). Such processes are generally driven by strategies to adapt to the harsh conditions in the atmosphere (e.g., rapid temperature and RH changes, thaw/freeze cycles, humidification/desiccation, UV exposure) (Hamilton and Lenton, 1998; Horneck et al., 1994; Joly et al., 2015; Setlow, 2007) or to limit their atmospheric residence time by initiating precipitation (Hernandez and Lindow, 2019). These processes include nutrient uptake by biodegradation (Khaled et al., 2020), bacteria cell generation that enhances particle size and surface area (Ervens and Amato, 2020), formation of biofilms (extracellular polymeric substances) which enables BAPs to form aggregates (Monier and Lindow, 2003, 2005; Morris et al., 2008; Sheng et al., 2010),
expression of ice-nucleating proteins (Joly et al., 2013; Kjelleberg and Hermansson, 1984), formation of biosurfactants that enhances water uptake (Hernandez and Lindow, 2019; Neu, 1996), desiccation that decreases size of BAPs (Barnard et al., 2013), formation of pigments (Pšenčík et al., 2004; Fong et al., 2001) enhancing light absorption, fungal spore germination (Ayerst, 1969) or formation of bacteria endospores (Enguita et al., 2003) that increases the $N_{\text{BAP}}$ and metabolism of cellular components (membranes, proteins, saccharides, osmolytes, etc) (Fox and Howlett, 2008; Xie et al., 2010). To date, the uncertainties introduced by these BAP ageing processes in the estimate of BAP radiative effects, their atmospheric residence time and distribution can only be assessed qualitatively due to the lack of comprehensive data. However, it may be expected that some of them lead to similar differences in BAP properties than differences between BAP types.

In our study, we give a brief overview of the BAP properties in Figure 1 and summarize which chemical, physical and biological processes may alter these properties (Section 2). By means of process models (Section 3), we explore in a simplistic way the relative importance of these BAP properties and ageing processes for their radiative effects depicted in Figure 1 (Section 4). The results of our sensitivity studies allow a ranking of the importance of the various BAP properties and processes in terms of their radiative impacts. In Section 5, we give some guidance on the need of future laboratory, field and model studies to more accurately describe the radiative effects, distribution and residence time of BAPs in the atmosphere.

2. Physicochemical properties and processes of BAPs

Literature data on physicochemical parameters of BAPs are summarized in Table 1. It is not our goal to repeat exhaustive reviews on these individual properties; for this, we refer to previous overview articles (Bauer et al., 2003; Coluzza et al., 2017; Deguillaume et al., 2008; Després et al., 2012; Fröhlich-Nowoisky et al., 2016; Hoose and Möhler, 2012; Huffman et al., 2020; Šantl-Temkiv et al., 2020). We rather aim at using characteristic orders of magnitude of these properties as input data to our process models (Section 3). Therefore, we only give a brief overview on the ranges and variability of these properties for different BAP types and due to various ageing processes.

2.1 BAP number size distribution parameters ($N_{\text{BAP}}$ and $D_{\text{BAP}}$)

The number concentration ($N_{\text{BAP}}$) of most BAP types is in the range of $0.001 \leq N_{\text{BAP}} \leq 0.1 \text{ cm}^{-3}$ (Table 1). The number concentration of bacteria is higher than that of fungal spores and pollen although the mass concentration of bacteria is lower (Burrows et al., 2009a; Heald and Spracklen, 2009; Hoese et al., 2010). $N_{\text{BAP}}$ can vary by about three orders of magnitude among different ecosystems, locations, seasons, and time of the day (Huffman et al., 2010, 2020; Matthias-Maser et al., 2000a, 2000b; Schumacher et al., 2013). The BAP diameter ($D_{\text{BAP}}$) covers a broad range of $0.01 \mu\text{m} \leq D_{\text{BAP}} \leq 100 \mu\text{m}$. This parameter usually refers to the mass equivalent diameter, which is the diameter of a sphere with the same mass as a non-spherical
BAP. The size depends on the types of BAPs, and on changes due to biological and physical processing. Viruses are reported to be the smallest BAP types ($0.01 \mu m \leq D_{\text{viruses}} \leq 0.3 \mu m$) while pollen is the largest ($5 \mu m \leq D_{\text{pollen}} \leq 100 \mu m$) (Table 1). Biological processing, such as cell generation, might increase the size of particles by producing secondary biological aerosol mass (Ervens and Amato, 2020; Sattler et al., 2001). Typical bacterial cell generation rates are in the range of 0.1-0.9 h$^{-1}$ (Ervens and Amato, 2020). Efficient generation in the atmosphere is assumed to be largely restricted to the time of cell exposure to liquid water (i.e., in-cloud). With an average atmospheric residence time of ~1 week (Burrows et al., 2009b) and an average in-cloud time fraction of ~15% (Lelieveld and Crutzen, 1990), it can be estimated that the generation time scale of bacteria cells in the atmosphere is on the order of ~20 h. Thus, for example, $D_{\text{bacteria}}$ may increase from 1 $\mu m$ to 2 $\mu m$ after one week in the atmosphere assuming a generation rate of 0.3 h$^{-1}$. Other rates, such as the cell growth, are usually much smaller (Marr, 1991; Middelboe, 2000; Price and Sowers, 2004; Sattler et al., 2001; Vrede et al., 2002), and thus, contribute less efficiently to a change in $D_{\text{BAP}}$. In addition, the formation of extracellular polymeric substances might lead to the formation of biofilms, which increase BAP size by forming agglomerate (Monier and Lindow, 2003, 2005). Agglomerate formation might be also described as a physical process, when BAPs (e.g. bacteria) attach to other particles (e.g. dust) (Després et al., 2012; Lighthart, 1997), which can result in particle sizes on the order of ~10 $\mu m$. Similarly, a biologically-driven physical processes might lead to enhancement of $N_{\text{BAP}}$ as it has been observed that pollen ruptures into fragments with diameter of 1-4 $\mu m$ during thunderstorms (Zhang et al., 2019).

### 2.2 Optical properties of BAPs: Complex refractive index ($m_{\text{BAP}} = n + ik$)

The scattering and absorption of particles are commonly described by the refractive index $m_{\text{BAP}}$ with real part ($n_{\text{BAP}}$) and imaginary parts ($k_{\text{BAP}}$) that depend on the chemical composition and wavelength of irradiation. Arakawa et al. (2003) reported $1.5 \leq n_{\text{BAP}} \leq 1.56$ and $3 \cdot 10^{-2} \leq k_{\text{BAP}} \leq 6 \cdot 10^{-4}$ for bacteria (Erwinia herbicola) in the wavelength range of 0.3-2.5 $\mu m$. Other groups found a broader range of $n$ and $k$ (Table 1) for different types of BAPs and irradiation wavelengths (Hu et al., 2019; Thrush et al., 2010). The imaginary part can vary by three orders of magnitude for different BAP types (Hu et al., 2019). Hill et al. (2015) showed that the refractive index of BAPs can be estimated based on the chemical composition. They reported $1.59 + i0.045$ for Bacillus vegetative cells at 0.266 $\mu m$. Also BAP shape (e.g. core-shell structure, hexagonal grids, and barbs), as it has been demonstrated for pollen, influences the optical properties (Liu and Yin, 2016). Due to the similarity of the molecular structure of organic macromolecules (e.g. proteins) and secondary organic aerosols (SOA), it can be likely assumed that might alter the BAP refractive index similar to that of SOA. Experimental results show $1.516 \leq n \leq 1.576$ and $0 \leq k \leq 0.013$ for fresh SOA; after nitration, the real part changed to $1.534 \leq n \leq 1.594$ and the imaginary part increased to $0.001 \leq k \leq 0.035$ (Liu et al., 2015; Moise et al., 2015).
Table 1. Physicochemical properties of various types of BAP and their changes due to physical, chemical and biological ageing processes based on literature data.

| BAP Types | Physicochemical properties |
|-----------|-----------------------------|
|           | Concentration $N_{BAP}$ (cm$^{-3}$) | Diameter $D_{BAP}$ (µm) | Hygroscopicity $\kappa$ | Complex refractive index $n_{BAP}$ ($\lambda$) = $n + ik$ | Surface tension $\sigma$ (mN m$^{-1}$) | IN active number fraction | Contact angle $\theta_{BAP}$ (°) |
| Bacteria  | 0.001-1 (1) 0.01-1.4 (2) | 1 (7) 0.6-7 (8) | 0.11-0.25 (11) | n: 1.5-1.56, k: 3·10$^{-5}$-6·10$^{-4}$ (17), n: 1.5-1.56, k: 0-0.04 (18), n: 1.25-1.87, k: 0-0.45 (19) | 25, 30, 55, 72 (20) | ~0.1%, ~1%, ~10% (21) | 32-34 (24), 4-20 (25), 28, 33, 44 (26) |
| Fungal spores | 0.001-0.01 (3) | 3-5 (4); 1-30 (5) | 0.03-0.073 (12), 0.036-0.048 (13), 0.05-0.1 (14), 0.08-0.17 (15), 0.14-0.24 (16) | ~100% (22, 23) | 14-30 (25), 15, 16.3 (28) |
| Fern spores | 10$^{-4}$ (4) | 1-30 (4) | 0.1 (6) 1-4 (6) | 0.01-0.3 (4), 0.04-0.2 (10) |
| Pollen fragments | 0.01 (4) | 0.01-0.3 (4), 0.04-0.2 (10) |

| Ageing processes of BAP | Physical ageing | Chemical ageing | Biological ageing |
|-------------------------|----------------|----------------|-------------------|
| Bacteria | Agglomeration: $\Delta N_{BAP} < 0$ (8) | Nitrification: $\Delta N_{BAP} \sim 0.02$, $\Delta k_{BAP} \sim 0.03$ (29). Nitrification: $\Delta \theta_{BAP} \sim 1°$ (26). pH changes: $\Delta \theta_{BAP} \sim 1.5°$ (26). | Biosurfactant production: $\sigma_{BAP} < 0$ (20). Biofilm formation: $\Delta D_{BAP} > 0$ (30). Endospore formation: $\Delta N_{BAP} > 0$ (31). Cell generation: $\Delta D_{BAP} > 0$ (32). Desiccation: $\Delta D_{BAP} < 0$ (33). Pigment formation (34, 35): $\Delta k > 0$. |
| Fungi | Rupture: $\Delta D_{BAP} < 0$, $\Delta N_{BAP} > 0$ (6) | Oxidation: $0.5 \leq \Delta \theta_{BAP} \leq 0.8°$ (28) | Biosurfactant production: $\sigma_{BAP} < 0$ (20). Germination: $\Delta N_{BAP} > 0$ (34). Desiccation: $\Delta D_{BAP} < 0$ (33). |

(1) Total bacteria, Tong and Lighthart et al., 1999; (2) Under haze conditions in Beijing, Wei et al., 2016; (3) Elbert et al., 2007; (4) Després et al., 2012; (5) blooming times, Huffman et al. 2010; (6) thunderstorm times, Zhang et al., 2019; (7) Burrows et al., 2009a; (8) Lighthart 1997; (9) Pöhlker et al., 2013; (10) Verreault et al., 2008; (11) Lee et al., 2002; (12) Pope et al. 2010; (13) Tang et al., 2019; (14) Chen et al., 2019; (15) Griffiths et al., 2012; (16) pollen kit, Prisle et al., 2019; (17) Arakawa et al., 2003; (18) Thrush et al., 2010; (19) Hu et al. 2019; (20) Renard et al., 2016; (21) T $\sim 10°$C, immersion freezing, *Pseudomonas syringae* bacteria, *Pseudoxanthomonas* sp., *Xanthomonas* sp., Joly et al., 2015; (22) deposition freezing for pollen, Diehl et al., 2001; (23) immersion and contact freezing for pollen, Diehl et al., 2002; (24) Hoose and Mohler, 2012; (25) Chen et al., 2008; (26) immersion freezing for *Pseudomonas syringae*, and *Pseudomonas fluorescens*, Attard et al., 2012; (27) immersion freezing for fungi, Kunert et al., 2019; (28) deposition freezing of silver birch and grey alder pollen, Gute and Abbatt, 2018; (29) nitrated SOA to represent nitrated BAP, Liu et al., 2015; (30) Morris et al., 2008; (31) Enguita et al., 2003; (32) Ervens and Amato, 2020; (33) Barnard et al., 2013; (34) Pšenčík et al., 2004; (35) Fong et al., 2001.
2.3 BAP Properties relevant for CCN activation

2.3.1 Hygroscopicity (κBAP) of BAPs

The hygroscopicity determines BAP’s hygroscopic growth factor (gf, as the ratio of wet to dry particle diameter) at subsaturated conditions and their CCN activity; it is usually expressed as the hygroscopicity parameter κ (Petters and Kreidenweis, 2007). Lee et al. (2002) reported gf = 1.16 for Bacillus subtilis bacteria and gf = 1.34 for Escherichia coli bacteria at RH ~85%. Based on these growth factors, κbacteria = 0.11 and κbacteria = 0.25 for these bacteria can be calculated. The hygroscopicity of pollen is similar: The κ value of intact pollen grains falls into the range of 0.03 ≤ κpollen ≤ 0.17 (Chen et al., 2019; Pope, 2010; Tang et al., 2019), in agreement with κ of pollen kits on the surface of pollen (0.14 ≤ κpollen ≤ 0.24) (Prisle et al., 2019).

2.3.2 Surface tension (σBAP) of BAPs

In most model studies that explore CCN activation, it is assumed that particles have a surface tension close to that of water (σwater = 72 mN m⁻¹). This assumption is likely justified under many conditions due to the strong dilution of internally mixed aerosol particles near droplet activation. There are numerous studies that postulate that surfactants in aerosol particles might influence the surface tension sufficiently to significantly change their CCN activity (Bzdek et al., 2020; Facchini et al., 1999; Lowe et al., 2019; Nozière et al., 2014). These surfactants are usually assumed to have natural sources such as the ocean surface (Gérard et al., 2019; Ovadnevaite et al., 2017). Another source of surfactants might be living microorganisms that produce biosurfactants which enhance surface hygroscopicity and decrease surface tension (Akbari et al., 2018). These biosurfactants might not only be associated with BAP themselves as they might be deposited on surfaces (e.g. leaves) where they can be taken up by other particles. Renard et al. (2016) reported that 41% of tested strains actively produced surfactant with σBAP < 55 mN m⁻¹ and 7% of tested strains can produce extremely efficient biosurfactants with σBAP < 30 mN m⁻¹. All of these tested strains were collected and isolated in cloud water samples. The efficient biosurfactants (σBAP < 45 mN m⁻¹) are mostly produced by Pseudomonas and Xanthomonas of bacteria (78%) and Udeniomyces fungi (11%). For the most efficient biosurfactants, we fit the following linear approximation based on their experiments:

\[
\sigma_{\text{BAP}} = 89.6 - 2.9 \cdot C_{\text{biosurf}} \quad \text{if } 6 \text{ mg L}^{-1} \leq C_{\text{biosurf}} \leq 22 \text{ mg L}^{-1} \quad (1)
\]

where σBAP is BAP surface tension in (mN m⁻¹) and Cbiosurf is the biosurfactant concentration in (mg L⁻¹). Higher and lower biosurfactant concentrations may be approximated with 25 mN m⁻¹ and 72 mN m⁻¹ for simplicity. Equation 1 implies that the concentration of biosurfactant on the surface is the same as in the bulk. Recent studies suggest that the surface concentration of surfactants is higher than the bulk concentration (Bzdek et al., 2020; Lowe et al., 2019; Ruehl et al., 2016). Thus, a smaller amount of
biosurfactants (‘critical micelle concentration’) than suggested by Equation 1 might be sufficient to significantly decrease $\sigma_{\text{BAP}}$. The biosurfactant concentration depends both on the dilution (amount of water) and on the mass fraction of biosurfactants in the particle. The mass fraction has not been determined for biosurfactants; however, other surfactants have been shown to contribute ~0.1% to the total particle mass (Gérard et al., 2019).

2.4 BAP Properties relevant for ice nucleation

2.4.1 IN active particle number fraction

In freezing experiments of pollen, it has been demonstrated that all particles freeze at sufficiently low temperatures, i.e. the IN active number fraction can be assumed as ~100%. Both condensation and immersion/contact freezing led to frozen fractions of 100% at T ~-18 °C (Diehl et al., 2001) and T ~-20 °C (Diehl et al., 2002), respectively. However, for bacteria such as *Pseudomonas syringae*, the maximum frozen fraction only reaches values of 0.1-10% at T ~-10 °C (Joly et al., 2013). This might be explained by the fact that not all of the bacteria cells express the same proteins even if they belong to the same species and the same population. It was observed that bacteria express more IN proteins under stress conditions (Kjelleberg and Hermansson, 1984), as a strategy to reach nutrients after destroying the cells of plants by freezing.

However, to date it is not fully understood why in lab experiments some of the bacteria cells show freezing behaviour while others from the same population do not and why individual cells show stochastic behaviour in repeated experiments (Lukas et al., 2020).

2.4.2 Contact angle between substrate and ice ($\theta_{\text{BAP}}$)

In agreement with previous studies, we base our discussion on the contact angle as a fit parameter in the classical nucleation theory (CNT) to parametrize the frozen fraction observed in experiments. Chen et al. (2008) reported $4^\circ \leq \theta_{\text{bacteria}} \leq 20^\circ$ and $14^\circ \leq \theta_{\text{pollen}} \leq 30^\circ$. Similarly, based on the measurements by Attard et al. (2012), we derived values of $28^\circ$, $33^\circ$, and $44^\circ$ for different types of bacteria. $\theta$ values for fungi based on the measurements by Kunert et al. (2019) are similar ($30^\circ \leq \theta_{\text{fungi}} \leq 33^\circ$). Gute and Abbatt (2018) performed deposition freezing experiments of pollen; based on their experiments, we fitted $\theta_{\text{pollen}} = 15^\circ$ for silver birch and $\theta_{\text{pollen}} = 16.3^\circ$ for grey alder. Hoose and Möhler (2012) reported the ice nucleation active surface site (INAS) density of various bacteria at -5 °C ($10^{2.5-10^{10}}$ m$^{-2}$). INAS implies that freezing occurs deterministically as opposed to stochastic freezing described by CNT. As the sensitivity of ice nucleation to time is generally small compared to other parameters (Ervens and Feingold, 2013), we fitted their data using CNT and obtained a range of $32^\circ \leq \theta_{\text{bacteria}} \leq 34^\circ$, consistent with other bacteria (Attard et al., 2012).

Chemical processes (e.g. nitration) can change the molecular surface of BAPs by e.g., adding nitro groups to tyrosine residues of proteins (Estillore et al., 2016), which can alter the IN activity. Attard et al. (2012)
measured the cumulative fraction of IN among a population of bacteria before and after nitration for 16-18 h. The residence time of aerosol particles in the atmosphere is from hours to weeks, which means that the experimental nitration times might be a realistic time scale. Based on these data, we calculated that the contact angle increased by ~1° after nitration for some bacteria. In contrast, Kunert et al. (2019) reported that protein nitration does not influence the cumulative fraction of IN for 65 species of fungi investigated.

In order to study the oxidation effect, Gute and Abbatt (2018) exposed pollen to OH radicals and measured the cumulative frozen fraction of pollen in terms of deposition freezing. We calculated that the contact angle increased by ~0.5° ≤ Δθpollen ≤ 0.8° after oxidation. While experimental conditions are often such that a large fraction of particles is nitrated or oxidized, respectively, only a small fraction of ambient proteins (~0.1%) have been found to be only nitrated (Franze et al., 2005). Attard et al., (2012) showed that a decrease of pH from 7.0 to 4.1, led to a decrease of the cumulative fraction of IN of P. syringae (32b-74) from 10⁻² to 10⁻⁸ at -4 °C. This change can be described by an increase of θ from 28.7° to 30.3° (Δθbacteria ~ 1.6°). P. syringae (CC0242), Snomax®, and P. fluorescens exhibited similar increases of Δθbacteria ~ 1.5° for the same change in pH.

3. Model description

3.1 Box model: Scattering/absorption of wet particles at RH < 100% calculated by Mie theory

A box model was used to simulate total scattering/absorption based on Mie theory (Bohren, 1983) for a constant aerosol distribution at different RH. Water uptake by particles is calculated based on Köhler theory. Mie theory is applied to calculate total scattering and absorption of the wet aerosol population as a function of D, N, and m at different wavelengths (λ). The input aerosol size distribution is based on ambient measurements by an ultraviolet aerodynamic particle sizer (UV-APS) in central Europe (Zhang et al., 2019) that cannot detect particles with D < 0.5 µm. At λ ≥ 300 nm, the particles with D > 3 µm interact with light by geometric scattering, rather than Mie scattering. Therefore, we intentionally only consider particles with diameters of 0.5 µm < D < 2.8 µm in 24 size classes to represent ambient aerosol particles relevant for our study with a concentration of Nother = 1.4 cm⁻³. We consider one additional BAP size class which has specific parameters (N_{BAP}, D_{BAP}, m_{BAP}, κ_{BAP}, σ_{BAP}) that are varied in the sensitivity studies.

Calculations are performed for RH of 10% and 90%, i.e. for different BAP growth factors. In a series of sensitivity studies (S_{qui} - S_{qui1}; Table 2), we explore the sensitivity of scattering and absorption to the N_{BAP}, D_{BAP}, κ_{BAP}, and m_{BAP} (m_{BAP} = n + ik). We not only compare model results for properties representing different BAP types (e.g. D_{bacteria} vs D_{fungal}), but also explore the ranges of property variation due to ageing processes of individual BAP types (e.g. ΔD_{bacteria}).
Table 2. Model sensitivity studies assume different physicochemical BAP parameters to investigate their effect on the optical properties (Section 3.1), CCN activation (Section 3.2) and ice nucleation (Section 3.3).

| Simulation | N_{BAP} (cm^{-3}) | D_{BAP} (µm) | n(λ)_{BAP}; k(λ)_{BAP} | RH | κ_{BAP} |
|------------|--------------------|---------------|------------------------|----|---------|
| S_{opt1}   | 0                  | -             | -                      | 10%| -       |
| S_{opt2}   | 0.01               | 1             | 1.5-1.56; 3-10^{-5}-6-10^{4} | 0.25|        |
| S_{opt3}   | 0.1                |               |                        |    |         |
| S_{opt4}   | 1                  |               |                        |    |         |
| S_{opt5}   | 0.1                | 2             |                        |    |         |
| S_{opt6}   | 3                  |               |                        |    |         |
| S_{opt7}   | 1                  | 2             |                        |    |         |
| S_{opt8}   |                    |               |                        | 90%| 0.03    |
| S_{opt9}   |                    |               |                        |    | 0.25    |
| S_{opt10}  |                    |               |                        |    | 0.25    |
| S_{opt11}  |                    |               |                        |    | 1.25-1.6; 0.001-0.1 | 10% |
| S_{opt12}  |                    |               |                        |    | 1.5-1.85; 0.08-0.5 |
| S_{opt13}  |                    |               |                        |    | 1.25-1.75; 0.06-0.32 |
| S_{opt14}  |                    |               |                        |    | 1.3-1.7; 0.05-0.1 |
| S_{opt15}  |                    |               |                        |    | 1.516-1.576; 0.0-0.013 |
| S_{opt16}  |                    |               |                        |    | 1.534-1.594; 0.001-0.035 |

Cloud condensation nuclei (CCN):

| Simulation | N_{BAP, IN} / N_{BAP} | RH | κ_{BAP} |
|------------|------------------------|----|---------|
| S_{CCN1}   | 0.5                    | 0.25 | 72     |
| S_{CCN2}   | 0.1                    | 0.03 |        |
| S_{CCN3}   | 0.05                   | 0.25 |        |
| S_{CCN4}   | 0.03                   | 0.25 |        |
| S_{CCN5}   | 0.03                   | 0.03 |        |
| S_{CCN6}   | 0.03                   | 0.03 |        |
| S_{CCN7}   | 0.5                    | 0.03 | 25     |
| S_{CCN8}   | 0.1                    | 0.03 |        |
| S_{CCN9}   | 0.05                   |      |        |

Ice nuclei (IN):

| Simulation | N_{BAP} (cm^{-3}) | D_{BAP} (µm) | Contact angle of ice germ (θ_{BAP}) | Cloud base temperature (°C) |
|------------|--------------------|---------------|-------------------------------------|----------------------------|
| S_{IN1}    | 1                  | 1             | 37°                                 | -8                         |
| S_{IN2}    | 0.01               |               |                                     |                            |
| S_{IN3}    | 0.001              |               |                                     |                            |
| S_{IN4}    | 0.01               | 2             |                                     |                            |
| S_{IN5}    |                    | 5             |                                     |                            |
| S_{IN6}    |                    | 1             | 4°                                  | -5.5                       |
| S_{IN7}    |                    |               | 20°                                 |                            |
| S_{IN8}    |                    |               | 40°                                 |                            |
| S_{IN9}    |                    |               | 38°                                 | -8                         |

Composition of other particles: 90% ammonium sulfate + 10% soot

Scattering/Absorption:

0.5 µm < D_{other} < 2.8 µm; N_{other} = 1.4 cm^{-3}; κ_{other} = 0.3; σ_{BAP} = σ_{other} = 72 mN m^{-1}

Cloud base temperature (°C): 46nm < D_{other} < 2.5 µm; N_{other} = 100 cm^{-3}

N_{BAP, IN} / N_{BAP}: 10%

θ_{other}: 80°
3.2 Adiabatic parcel model

3.2.1 CCN activation in warm clouds

An adiabatic parcel model was applied to simulate the formation of warm clouds (Ervens et al., 2005; Feingold and Heymsfield, 1992). The activation of an aerosol population to cloud droplets is described as a function of $N$, $D$, $\kappa$, and $\sigma$. The dry aerosol size distribution covers a size range of $5 \text{ nm} < D_{\text{other}} < 7.7 \text{ \mu m}$ with $N_{\text{other}} = 902 \text{ cm}^{-3}$. Similar to the studies on optical properties (Section 3.1), we assume that one aerosol size class is composed of biological material for which we vary $D_{\text{BAP}}$, $\kappa_{\text{BAP}}$, and $\sigma_{\text{BAP}}$ to explore the role of differences in BAP type and ageing processes on cloud droplet activation ($S_{\text{CCN1}}$–$S_{\text{CCN9}}$, Table 2).

3.2.2 Ice nucleation in mixed-phase clouds

The adiabatic parcel model as used for the CCN calculations was extended by the description of immersion freezing based on classical nucleation theory (Ervens et al., 2011). At each model time step (1 second), the frozen fraction of BAP is calculated; if 1% or more of the IN size class are predicted to freeze in a given time step, a new size class of ice particles is generated in the model, for which ice growth is described. We consider an aerosol size distribution with $46 \text{ nm} < D_{\text{other}} < 2.48 \text{ \mu m}$ in nine size classes and $N_{\text{other}} = 100 \text{ cm}^{-3}$ and one additional BAP size class, which is the only one that includes potentially freezing IN under the model conditions. Similar to the analysis by Ervens et al. (2011), we compare the evolution of the ice liquid water contents (IWC and LWC) expressed in mass fractions [%] whereas 100% corresponds to the total water (ice + liquid + vapor) mixing ratio that is constant under the adiabatic model conditions. Input values of $D_{\text{BAP}}$, $N_{\text{BAP}}$, and $\theta_{\text{BAP}}$ are varied in simulation $S_{\text{IN1}}$–$S_{\text{IN9}}$ (Table 2).

4. Results and discussion

4.1 Sensitivity of optical properties at subsaturated conditions (RH < 100%) to BAP properties

4.1.1 Influence of concentration (NBAP) and diameter (DBAP) on scattering and absorption

As explained in Section 3.1, in the sensitivity studies of optical properties, we consider only particles with $D$ in the same range as $\lambda$ so that scattering and absorption can be calculated by Mie theory. As ambient aerosol particles also include smaller and larger particles, our conclusions on BAP direct radiative effects should be regarded as the upper limit on total scattering and absorption.

In Figure 2, we compare the total scattering coefficient for a case without BAP ($N_{\text{BAP}} = 0$, $S_{\text{opt1}}$) to that predicted for $N_{\text{BAP}} = 0.01 \text{ cm}^{-3}$ ($S_{\text{opt2}}$), $N_{\text{BAP}} = 0.1 \text{ cm}^{-3}$ ($S_{\text{opt3}}$) and $N_{\text{BAP}} = 1 \text{ cm}^{-3}$ ($S_{\text{opt4}}$). At a typical low number concentration of $N_{\text{BAP}} = 0.01 \text{ cm}^{-3}$, the effect on total scattering coefficient is negligible. At $N_{\text{BAP}} = 0.1 \text{ cm}^{-3}$ the total scattering coefficient increases by 15-18% at $\lambda = 0.3-1.5 \text{ \mu m}$ although the number fraction of BAP is only 6%. We also use a higher concentration ($N_{\text{BAP}} = 1 \text{ cm}^{-3}$) to explore the maximum effect as BAP concentrations of this order of magnitude have been observed under haze conditions (Wei et al., 2016).
Total scattering coefficient changes by a factor of 0.5-2 depending on $\lambda$. Note that the concentration of other particles ($N_{\text{other}}$) would usually increase under haze conditions while we keep $N_{\text{other}}$ as a constant in the above model (1.4 cm$^{-3}$); therefore, the predicted increase of scattering coefficient is clearly the upper limit of the BAP effect. The absorption coefficient of the total aerosol population does not change.

**Figure 2.** Influence of BAP concentration on total scattering coefficient. The detailed input parameters can be found in Table 2. The black line, red line, blue line, and brown line correspond to $S_{\text{opt}1}$, $S_{\text{opt}2}$, $S_{\text{opt}3}$, and $S_{\text{opt}4}$ in Table 2, respectively.

$D_{\text{BAP}}$ also affects the scattering coefficient of the aerosol population significantly (Figure 3). $D_{\text{BAP}} = 1 \mu m$ ($S_{\text{opt}3}$) and $D_{\text{BAP}} = 2 \mu m$ ($S_{\text{opt}5}$) can be considered to represent different BAP types such as bacteria and fungi, respectively, or an aged bacteria cell that has undergone processing by cell generation (Ervens and Amato, 2020). For these assumptions, the scattering coefficient increases depending on $\lambda$, with the largest changes of 73-100% at $\lambda > 1.5 \mu m$ when $D_{\text{BAP}}$ increases from 1 $\mu m$ to 2 $\mu m$ ($S_{\text{opt}5}$). Larger BAP ($D_{\text{BAP}} = 3 \mu m$, $S_{\text{opt}6}$) such as pollen fragments show lead to an increase in the scattering coefficient by a factor of 1.4-4.7 depending on $\lambda$. The absorption coefficient of the aerosol population remains nearly the same.

The results in Figure 3 clearly show that the size of BAP needs to be known in order to assess their optical properties. Even a relatively small variation in particle diameter from 1 to 2 $\mu m$ due to different BAP types or to cell size changes ($\Delta D_{\text{BAP}}$) might lead to change in scattering coefficient by 8-100 % depending on $\lambda$. Given that the diameter ($D_{\text{BAP}}$) might vary by four orders of magnitude among different BAP types, our analysis shows that different sizes for the various BAP types need to be taken into account if their optical properties are evaluated.
Figure 3. Influence of BAP diameter on (a) scattering coefficient and (b) absorption coefficient of total particles. The detailed input parameters can be found in Table 2. The black line, red line, blue line, and brown line correspond to Sopt1, Sopt3, Sopt5, and Sopt6, respectively.

In our model studies, we make the simplistic assumption of spherical BAP particles. Electron scanning microscopic imaging has shown that BAP are not spherical but exhibit a variety of different shapes (Valsan et al., 2015; Wittmaack et al., 2005). The consequences of the assumptions of spherical versus non-spherical pollen on the derivation of optical properties at wavelength of 0.65 µm have been recently discussed (Liu and Yin, 2016). The extinction efficiency (sum of scattering efficiency and absorption efficiency) can vary by a factor of one to three for small pollen with $D < 4 \mu m$. For larger pollen with $D > 5 \mu m$, the extinction efficiency varies by ~25\% (Liu and Yin, 2016). While we do not explore sensitivities of BAP geometry, it may be postulated that under atmospheric conditions, i.e. when BAP are wet, they are more spherical than under the experimental dry conditions, and thus effects due to non-sphericity may be reduced.

4.1.2 Influence of hygroscopicity ($\kappa_{BAP}$) and surface tension ($\sigma_{BAP}$) on scattering and absorption

As discussed in Section 2.3, $g f_{BAP}$ might vary depending on BAP hygroscopicity ($\kappa_{BAP}$) and surface tension ($\sigma_{BAP}$). Figure 4 shows the influence of $\kappa$ on scattering and absorption at RH of 10\% ($S_{opt7}$, $S_{opt8}$) and 90\% ($S_{opt9}$, $S_{opt10}$). At RH = 10\% ($S_{opt7}$, $S_{opt8}$), the influence of BAP on scattering coefficient of total particles is small (\(< 19\%) and the influence on absorption coefficient is negligible. Assuming $\kappa = 0.25$ ($S_{opt10}$) instead of $\kappa = 0.03$ ($S_{opt9}$), leads to an increase of the scattering coefficient by 17-90\% at RH = 90\%. Also the absorption coefficient increases by \~40\% at $\lambda > 2 \mu m$. It can be concluded that the importance of $\Delta \kappa$ increases for higher RH as under these conditions BAP hygroscopic growth is most efficient.
In addition to hygroscopicity ($\kappa_{\text{BAP}}$), we explore the importance of biosurfactants which decrease surface tension of particles ($\sigma_{\text{BAP}}$). A lower surface tension leads to a reduced particle curvature which, in turn, enhances the water uptake. Numerically, this is expressed in the Köhler equation, which can be generally expressed as Equation 2:

$$s = \exp\left(\frac{A(\sigma)}{D_{\text{wet}}} - \frac{B(\kappa)}{D_{\text{wet}}}\right)$$

(2)

where $s$ is the equilibrium water vapor saturation ratio, $D_{\text{wet}}$ the wet particle diameter, the first term in the parentheses is the Kelvin (curvature) term which is a function of surface tension ($\sigma_{\text{BAP}}$) following Equation 3 and the second term is the Raoult (solute) term which can be parameterized by $\kappa_{\text{BAP}}$ (Rose et al., 2008) following Equation 4:

$$\text{Kelvin term} = \frac{A(\sigma)}{D_{\text{wet}}} = \frac{4\sigma_{\text{sat}}M_{\text{sat}}}{\mu \rho \omega RT D_{\text{wet}}}$$

(3)
\[ \text{Raoult term} = \frac{\sigma (\kappa)}{D_{\text{wet}}^3} = - \ln \frac{D_{\text{wet}}^3 - D_s^3}{D_{\text{wet}}^3 - D_s^3 (1 - \kappa)} \]  

(4)

where \( \sigma_{sol} \) is surface tension of solution droplet (72 mN m\(^{-1}\)); \( M_\omega \) is molar mass of water (18 g mol\(^{-1}\)); \( \rho_\omega \) is density of water (1 g cm\(^{-3}\)); \( R \) is the universal gas constant (8.31 \cdot 10^7 g cm\(^2\) s\(^{-2}\) K\(^{-1}\) mol\(^{-1}\)); \( T \) is the absolute temperature (K); \( D_{\text{wet}} \) is droplet diameter (cm); and \( D_s \) is the diameter of dry solute particle (cm).

The comparison of the two dimensionless terms shows that in most of the cases, the Raoult term exceeds the Kelvin term by at least one order of magnitude. Only for very small BAP, i.e. representative for viruses or bacteria fragments (Section 2.1), the curvature term significantly influences \( s \) (Figure 5). Based on this analysis, we can conclude that (bio)surfactants likely do not have a significant impact on the hygroscopic growth of BAP. A coating with surfactants might slow down the kinetics of the water uptake by particles (Davidovits et al., 2006). However, since the growth time scales of particles at RH < 100% are usually relatively long, the impact of surfactants on the time scale to reach equilibrium sizes is likely small, leading to a small importance of the effect of surfactant on water uptake and the corresponding optical properties.

**Figure 5.** Kelvin term as a function of surface tension (\( \sigma_{BAP} \)) for the \( \sigma \) range as found for BAP (left axis; solid lines). Raoult term as a function of hygroscopicity (\( \kappa_{BAP} \)) for the range of \( \kappa \) as found for BAP (right axis; dashed lines).

### 4.1.3 Influence of complex refractive index (\( m_{\text{BAP}} = n + ik \)) on scattering and absorption

The complex refractive index of BAP can be explained by their building blocks of various functional groups (Hill et al., 2015). Hu et al. (2019) have measured the complex refractive indices of 12 types of BAP including bacteria, pollen, and spores. Here we use *Bacillus subtilis* bacteria and *Lactobacillus acidophilus*.
bacteria, *Aspergillus oryzae* fungal spores, and lotus pollen as representative BAP types (*Table 1*) to show how the refractive index of different BAP might affect scattering and absorption coefficients of total particles. The scattering coefficient can vary by a factor of two and the absorption coefficient by a factor of five, depending on the wavelength with the largest effects at $\lambda > 2 $ µm (*Figure 6*).

*Figure 6.* The influence of different types of BAP on (a) the scattering coefficient and (b) absorption coefficient of total particles. The black, red, blue, and brown lines correspond to $S_{\text{opt11}}$, $S_{\text{opt12}}$, $S_{\text{opt13}}$, and $S_{\text{opt14}}$, respectively. All of other parameters are assumed to be equal (i.e. $D_{\text{BAP}}$, $N_{\text{BAP}}$, $\kappa_{\text{BAP}}$ and RH).

In addition to the variability in refractive index due to BAP type, chemical processing of the macromolecules at the BAP surface might modify the refractive index. It has been shown that nitration of SOA, i.e. the addition of a nitro group, leads to the formation of brown carbon (Moise et al., 2015). Qualitatively, it has been demonstrated that proteins can be nitrated, similar to SOA compounds (Shiraiwa et al., 2012). Due to the lack of data on $\Delta m$ for nitrated proteins in BAP, we assume a similar change in the refractive index as in SOA ($S_{\text{opt13}}$ and $S_{\text{opt14}}$). The scattering coefficient can change by up to 20% and the absorption coefficient by a factor of three at $\lambda = 0.42 $ µm (*Figure 7*). Thus, the variability in scattering/absorption properties of BAP due to $\Delta m$ caused by nitration is likely smaller than due to $\Delta m$ caused by different BAP types. The assumptions on $\Delta m$ made for the simulations shown in *Figure 7* are likely an overestimate of the chemical processing of proteins as BAP constituents since (1) experimental conditions are often optimized so that a large fraction of particles is nitrated (Liu et al., 2015) as opposed to ~0.1% of nitrated proteins observed in the atmosphere (Franze et al., 2005), (2) we assume nitration to occur over the whole residence time of particles in the atmosphere while proteins can be nitrated only under conditions of sufficiently high NOx levels (Shiraiwa et al., 2012), and (3) a rather high concentration of $N_{\text{BAP}} = 1 $ cm$^{-3}$ is considered.
4.1.4 Estimate of change of radiative forcing introduced by BAP

In order to give an estimate of the local radiative forcing due to BAPs, we applied the same approach as Dinar et al. (2007). The radiative forcing efficiency (RFE, i.e. radiative forcing per unit optical depth) at 390 nm can be calculated as:

$$ RFE = S_{con} D_{ten} (1 - A_{cld}) T_{atm}^2 (1 - R_{sfc})^2 \left[ 2 R_{sfc} \left( \frac{1-\omega}{(1-R_{sfc})^2} - \beta \omega \right) \right] $$

where $S_{con}$ is the solar constant (1370 W m$^{-2}$); $D_{ten}$ is the fractional day length (0.5); $A_{cld}$ is the fractional cloud cover (0.6); $T_{atm}$ is the solar atmospheric transmittance (0.76), and $R_{sfc}$ is surface albedo (0.15); $\omega$ is the single scattering albedo (SSA), which is the ratio of scattering coefficient to extinction coefficient; $\beta$ is average upscatter fraction, which can be calculated as:

$$ \beta = 0.082 + 1.85b - 2.97b^2 $$

where $b$ is the ratio of backscattering to scattering coefficient, $g$ is the asymmetry factor which is assumed as 0.65 as an average of ambient measurements (~0.59-0.72 (Andrews et al., 2006)). The calculated RFE are listed in Table 3 for some of the model results of the simulations listed in Table 2. The first row is the reference with internally mixed ammonium sulfate/soot particles only while BAP are absent. With a typical concentration of $N_{BAP} = 0.01$ cm$^{-3}$ ($S_{opt2}$), SSA increases both at $\lambda = 390$ nm and at $\lambda = 532$ nm, which means that BAP have a net cooling effect of $\Delta RFE = -0.22$ W m$^{-2}$ at $\lambda = 390$ nm and $\Delta RFE = -0.15$ W m$^{-2}$ at $\lambda = 532$ nm, respectively.

Figure 7. The influence of protein nitration on (a) the scattering coefficient and (b) absorption coefficient of total particles. The black and red correspond to $S_{opt15}$ and $S_{opt16}$, respectively.
Table 3. Radiative forcing efficiency (RFE) at 390 nm and 532 nm calculated based on Equations 7 and 8 for the Mie sensitivity studies. Some typical conditions are shown here to demonstrate the influence of various properties of BAP.

| Simulation          | SSA | RFE (W m⁻²) | ΔRFE (W m⁻²) | SSA | RFE (W m⁻²) | ΔRFE (W m⁻²) |
|---------------------|-----|-------------|--------------|-----|-------------|--------------|
|                     |     | 390 nm      | 532 nm       |     |             |              |
| S_{opt1} (without BAP, reference) | 0.643 | -0.5 | - | 0.728 | -6.84 | - |
| S_{opt2} (NBAP = 0.01 cm⁻³, DBAP = 1 µm) | 0.664 | -0.72 | -0.22 | 0.73 | -6.99 | -0.15 |
| S_{opt3} (NBAP = 0.1 cm⁻³, DBAP = 1 µm) | 0.668 | -2.36 | -1.86 | 0.747 | -8.26 | -1.42 |
| S_{opt4} (NBAP = 0.1 cm⁻³, DBAP = 2 µm) | 0.738 | -7.59 | -7.09 | 0.791 | -11.54 | -4.68 |
| S_{opt5} (NBAP = 0.1 cm⁻³, DBAP = 2 µm, Δm_{fung} fungal spores) | 0.535 | 7.56 | 8.06 | 0.56 | 5.69 | 12.53 |
| S_{opt6} (NBAP = 0.1 cm⁻³, DBAP = 2 µm, Δm_{fung} fresh BAP) | 0.852 | -16.1 | -15.6 | 0.929 | -21.83 | -14.99 |
| S_{opt7} (NBAP = 0.1 cm⁻³, DBAP = 2 µm, Δm_{fung} aged BAP) | 0.692 | -4.15 | -3.65 | 0.891 | -19 | -12.16 |

Note that the RFE values in Table 3 only represent radiative forcing of a small range of particle sizes and a constant composition and number concentration of other particles; however, the relative differences (ΔRFE) are meaningful and allow evaluating the relative importance of the various BAP parameters (NBAP, DBAP, Δm_{BAP}) in terms of the direct radiative effect. A decrease in RFE implies less absorption, and thus more cooling of atmosphere (Dinar et al., 2007). ΔNBAP (S_{opt3}) and ΔDBAP (S_{opt4}) have a significant influence on ΔRFE. In addition, ΔRFE at λ = 390 nm is higher than that at λ = 532 nm, implying the increasing importance of BAP in the UV range.

When fungal spores are considered instead of bacteria (i.e., m = n + ik is changed), SSA decreases and RFE even changes from a negative to a positive value (S_{opt13}), resulting in predicted RFE of 7.56 W m⁻² at λ = 390 nm and 5.69 W m⁻² at λ = 532 nm, respectively. This might be explained by the strong light absorption (very high k) of Aspergillus oryzae fungal spores. Generally, the imaginary part k can vary by three orders of magnitude between different types of BAP (Table 1), which makes both the sign and the absolute value of the radiative effects of BAP uncertain. Compared to the fresh BAP (S_{opt5}), the cooling effect of nitrated BAP (S_{opt6}) decreases, which can be explained by the increase of k for nitrated BAP due to the formation of brown carbon. These sensitivity studies demonstrate the significant effect of BAP in direct radiative forcing of supermicron particles. The properties of BAP (e.g. NBAP, DBAP, m_{BAP}) can vary depending on species of BAP and ageing processes. The Largest ΔRFE are caused by Δm_{BAP}.
4.2 Sensitivity of CCN activity to BAP properties

4.2.1 Influence of BAP concentration (NBAP) and diameter (DBAP) on CCN activation

NBAP is low compared to the total CCN concentration (Chow et al., 2015; Sun and Ariya, 2006). The upper limit NBAP is on the order of ~1 cm⁻³ (Table 1) while the number concentration of CCN is usually in the range of 10s to 1000s cm⁻³ (Ervens et al., 2010). The highest NBAP was found under haze conditions together with very high total particle concentrations: During haze days in Beijing, NBAP can reach up to ~1.4 cm⁻³ (Wei et al., 2016) when NCCN ~10³ cm⁻³ (Gunthe et al., 2011). Thus, the ratio of NBAP / Ntot or NBAP / NCCN is likely small, i.e. in a range of 0.01-0.14%, independent of location. While such a marginal increase in the number concentration of cloud droplets does not lead to an observable change in cloud properties, the properties related to the CCN activation of BAP should be considered being more important for biological reasons, i.e. for BAP to be surrounded by water and the significant modification of the atmospheric residence time of BAP that is consequently changed by the transport and precipitation in clouds.

The critical saturation $s_c$ can be used as a measure to estimate whether a particle will be activated into a cloud droplet (Rose et al., 2008):

$$s_c = \exp\left(\frac{4A^2}{27\kappa D_c^2}\right)$$

(7)

where $A$ can be found at Equation 5, $\kappa$ is hygroscopicity, and $D_c$ (cm) is mass equivalent diameter of dry solute particle. Applying this equation, one finds that for particles with $D_{BAP} = 0.01–10 $ µm, the critical supersaturations ($S_c = (s_c - 1) \cdot 100\%$) are in a broad range of 0.0007%-24% (assuming $\kappa = 0.03; \sigma = 72$ mN m⁻¹). For large BAP with $D_{BAP} > 0.5 $ µm, the critical supersaturations $S_c$ is smaller than 0.062%. Typical supersaturations ($S_{env}$) in stratocumulus and convective cumulus clouds are in the range of ~0.1-0.5% and ~0.5-1%, respectively (Pruppacher and Klett, 1997). Comparison to $S_{env}$ shows that most BAP ($D_{BAP} > 0.5 $ µm) are likely activated in clouds as their $S_c$ are significantly smaller than $S_{env}$ in clouds.

4.2.2 Influence of the hygroscopicity ($\kappa_{BAP}$) and surface tension ($\sigma_{BAP}$) on CCN activation

Figure 8a shows the range of $S_c$ for the $\kappa$ values shown in Table 2 for the smallest BAP with $D_{BAP} = 500 $ nm, 100 nm, and 50 nm. For $D_{BAP} = 500 $ nm, $S_c$ is 0.02% ($\kappa_{BAP} = 0.25; S_{CCN1}$) or 0.06% ($\kappa_{BAP} = 0.03; S_{CCN2}$), which are both below typical $S_{env}$ in clouds. Only for smaller BAP such as bacterial fragments or viruses with $D_{BAP} = 100 $ nm ($S_{CCN1, S_{CCN2}}$), $S_c$ changes from 0.24% ($S_{CCN3}$) to 0.69% ($S_{CCN4}$) when $\Delta\kappa_{BAP} = 0.22$. For even smaller $D_{BAP} (50$ nm), $S_c$ increases from 0.68% ($S_{CCN5}$) to 1.97% ($S_{CCN6}$) when $\Delta\kappa_{BAP} = 0.22$. Thus, only for fairly small BAP, the hygroscopicity $\kappa_{BAP}$ may impact their CCN activation.

Overlaid on the vertical lines for $S_c$ in Figure 8a are $S_{env}$ in the cloud as calculated in our parcel model for different updraft velocities ($w = 10 $ cm s⁻¹, 100 cm s⁻¹, and 300 cm s⁻¹). The sensitivity of CCN properties to
updraft velocity and $S_{\text{env}}$ has been discussed in numerous previous studies, e.g., Ervens et al. (2005, 2007). Figure 8a corroborates the conclusions from these previous studies that the variation of the $\kappa$ over wide ranges only introduces a small change in the CCN activity and in cloud properties (e.g., drop number concentration, LWC) and that particle composition is most important in clouds with low updraft velocities.

**Figure 8.** Comparison of $S_{\text{env}}$ to $S_c$ of BAP with (a) different $\kappa$ or (b) different $\sigma$. Model details can be found in Table 2.

Similar to $S_c$ ranges due to different $\kappa_{\text{BAP}}$ values, we compare in Figure 8b predicted $S_c$ ranges due to different values of $\sigma_{\text{BAP}}$ for high biosurfactant concentrations (when mass fraction of surfactants to total particle mass $> 0.1\%$, $\sigma_{\text{BAP}} = 25 \text{ mN m}^{-1}$) to those predicted for very low surfactant concentrations ($\sigma_{\text{BAP}} = 72 \text{ mN m}^{-1}$). For BAP with $D_{\text{BAP}} = 500 \text{ nm}$, $S_c$ changes from $0.01\%$ ($S_{\text{CCN} \text{7}}$, $\sigma_{\text{BAP}} = 25 \text{ mN m}^{-1}$) to $0.06\%$ ($S_{\text{CCN} \text{2}}$, $\sigma_{\text{BAP}} = 72 \text{ mN m}^{-1}$). As discussed before, these large BAP will be likely all activated in clouds and the small difference in $S_c$ introduced by $\Delta \sigma$ does not cause a difference in their CCN ability. For smaller BAP, such as bacterial fragments or viruses with $D_{\text{BAP}} = 100 \text{ nm}$, $S_c$ changes from $0.14\%$ ($S_{\text{CCN} \text{9}}$) to $0.69\%$ ($S_{\text{CCN} \text{6}}$) when $\Delta \sigma_{\text{BAP}} = 47 \text{ mN m}^{-1}$. When $D_{\text{BAP}}$ further decreases to $50 \text{ nm}$, $S_c$ changes from $0.4\%$ ($S_{\text{CCN} \text{9}}$) to $1.97\%$ ($S_{\text{CCN} \text{6}}$) when $\Delta \sigma_{\text{BAP}} = 47 \text{ mN m}^{-1}$. Therefore, the effect of biosurfactant needs to be considered for small BAP in terms of CCN activity if a sufficiently large mass fraction of strongly surface-active biosurfactant is present. Note that the assumption of $\sigma_{\text{BAP}} = 25 \text{ mN m}^{-1}$ in Figure 8b likely represents an overestimate as most biosurfactants exhibit a range of $30 \text{ mN m}^{-1} < \sigma_{\text{BAP}} < 55 \text{ mN m}^{-1}$ (Renard et al., 2016). In addition, the biosurfactant concentration, and thus the surface tension according to Equation 1, depends...
on the mass fraction of biosurfactants in the BAP, the growth factor and on diameter of BAP. If the mass fraction is very low, $\sigma_{BAP} = 72 \text{ mN m}^{-1}$; when the mass fraction of biosurfactants approaches $\sim$0.1%, $\sigma_{BAP}$ might be as low as 25 mN m$^{-1}$. Typical surfactant mass concentrations are on the order of $\sim$0.1% (Gérard et al., 2019); mass fractions for specific biosurfactants have not been determined yet. Such low mass fraction implies that only a few ($< 10–100$) surfactant molecules (with a molecular weight $M \sim 1000 \text{ g mol}^{-1}$) are present on submicron particles and/or that only a fraction of particles is completely covered by surfactants and thus exhibits a reduced surface tension. While biosurfactants might be also taken up by other particles while they reside on surfaces (soil, vegetation) where BAP were active, our conclusions also hold for such particles. Our sensitivity studies show once more that under dynamic conditions in clouds buffering reduces the feedbacks of particle composition on supersaturation (Ervens et al., 2005; Stevens and Feingold, 2009). Therefore, previous estimates of surfactant effects on cloud properties that are based on a simplified assumption of equilibrium conditions in clouds (Facchini et al., 1999), led to an overestimate of the role of surfactants on CCN.

We conclude that the mass concentration of biosurfactants needs to be quantified in order to better explore the biosurfactant effect on CCN activation of small particles. Given that the surface concentration of surfactants is likely higher than the bulk concentration (Bzdek et al., 2020; Lowe et al., 2019; Ruehl et al., 2016) as assumed here, even a smaller mass fraction of biosurfactants than calculated by Equation 1 might be sufficient to decrease the surface tension of small aqueous BAP and the corresponding critical supersaturation. However, also for the concept of surface partitioning of biosurfactants, rather than for a bulk concentration, our conclusions hold true on the limited impact of surface tension suppression on CCN activation of supermicron BAPs.

4.3 Sensitivity of mixed-phase cloud evolution to BAP properties

4.3.1 Influence of BAP concentration ($N_{BAP}$) and diameter ($D_{BAP}$) on ice nucleation

$N_{BAP}$ is on the same order of magnitude as that of total IN in some regions and at temperatures $T > \sim$10$^0$C (Pratt et al., 2009; Prenni et al., 2009), which makes BAP play an important role in mixed-phase clouds. Especially, at these relatively high temperatures, some bacteria and fungi species can nucleate ice while other particles cannot, and therefore $N_{BAP, IN}/N_{IN}$ is $\sim$100%. Figure 9a shows the change of ice water content (IWC, solid lines) and liquid water content (LWC, dashed lines) in a mixed-phase cloud ($S_{IN1}$, $S_{IN2}$, $S_{IN3}$). Above an IWC of $\sim$3%, ice particles start growing at the expense of liquid water (Bergeron-Findeisen-Process) ($S_{IN1}$). At lower $N_{BAP} \sim 0.01 \text{ cm}^{-3}$($S_{IN2}$), the onset of the Bergeron-Findeisen-Process starts slightly later. With $N_{BAP} \sim 0.001 \text{ cm}^{-3}$($S_{IN3}$), both IWC and LWC are predicted to increase simultaneously throughout the whole cloud, i.e. the Bergeron-Findeisen-Process is not initiated and cloud glaciation does not take place.
In **Figure 9b**, we compare model results for simulations S\textsubscript{IN4} and S\textsubscript{IN5} in order to explore the effect of D\textsubscript{BAP}. With larger BAP size such as D\textsubscript{BAP} = 2 \, \mu m (S\textsubscript{IN4}) or D\textsubscript{BAP} = 5 \, \mu m (S\textsubscript{IN5}), ice formation starts earlier in the cloud, but the onset of the Bergeron-Findeisen process occurs at approximately the same temperature as for smaller D\textsubscript{BAP} because of the feedbacks of IWC and LWC on the supersaturations in the cloud and vice versa. In agreement with previous sensitivity studies (Ervens et al., 2011; Ervens and Feingold, 2013), these results confirm that the influence of D on the IN activity is relatively small (**Figure 9**). Based on these trends, it can be also concluded that processes that change the BAP size (e.g. \Delta D\textsubscript{BAP} by cell generation) are not critical to be included in models to represent the variability of IN property effect on mixed-phase clouds.

![Figure 9](https://doi.org/10.5194/acp-2020-781)

**Figure 9.** Ice water content (IWC, solid lines) and liquid water content (LWC, dashed lines) as a function of (a) N\textsubscript{BAP} and (b) D\textsubscript{BAP}. Details on the simulations can be found in **Table 2**.

### 4.3.2 Influence of the contact angle (\theta\textsubscript{BAP}) on ice nucleation

Different types of BAPs exhibit a wide range of contact angles of 4° < \theta\textsubscript{BAP} < 44° (**Table 1** and **Section 2**). As shown in **Figure 10**, different BAP types that have \theta\textsubscript{BAP} of 4° or 20°, respectively, lead to a difference in temperature, at which the Bergeron-Findeisen process occurs, by \Delta T \approx 0.6 \, ^\circ \text{C}. For BAP with even higher \theta\textsubscript{BAP} (40°), the Bergeron-Findeisen process occurs even at a lower temperature (\Delta T \approx 3.3 \, ^\circ \text{C}).

As discussed in **Section 2**, chemical (e.g., nitration, oxidation, adjustments due to pH) or physical processing of IN surfaces might lead to \Delta \theta\textsubscript{BAP} \approx 1°. In **Figure 10d**, we show the resulting change in IWC and LWC evolution by comparing S\textsubscript{IN2} and S\textsubscript{IN9}. It is clear that even such a small change in \theta can cause a significant difference in the IWC and LWC evolutions. The temperature, at which the Bergeron-Findeisen process occurs differs by \Delta T \approx 1.3 \, ^\circ \text{C}. These results suggest that a small change of contact angle due to different types of BAP or due to processing (\Delta \theta) might affect the Bergeron-Findeisen process significantly. We only exemplarily explore \Delta \theta for nitration based on the experiments by Attard et al. (2012). In the same study, it
was found that $\Delta \theta$ is $\sim 1.5^\circ$ for bacteria such as *Pseudomonas syringae* when the cells were exposed to solutions of pH 7.0 and 4.1 at temperatures of $T > -10 ^\circ C$.

![Graphs showing ice water content (IWC) and liquid water content (LWC) as a function of $\theta_{BAP}$](image)

**Figure 10.** Ice water content (IWC, solid lines) and liquid water content (LWC, dashed lines) as a function of $\theta_{BAP}$. Even when the contact angle increases by 1°, the initiation of Bergeron-Findeisen process might be influenced significantly.

Similar exercises could be done for differences in $\theta$ due to other processes, such as the oxidation of pollen that lead to $\Delta \theta \sim 1.5^\circ$ at $T \sim -39 ^\circ C$ (Gute and Abbatt, 2018). However, at this much lower temperature, the sensitivity of the frozen fraction to $\Delta \theta$ decreases (Ervens and Feingold, 2013). Overall, it can be concluded that chemical processing of bacteria or other BAP that freeze at relatively high temperatures in the atmosphere for extended periods of time might sufficiently alter their surface to induce a significant change in their IN ability.
5. Conclusions

Based on our model sensitivity studies, we can rank the importance of the various parameters and processes of BAPs shown in Figure 1 in terms of their radiative effects: The increasing importance and sensitivity are summarized in Figure 11.

![Schematic of the importance of BAP in the climate system and the sensitivity of radiative effect to BAP properties. The bottom arrow shows the increasing importance of BAP in CCN, scattering/absorption, and IN. The left arrow indicates the increasing sensitivity to BAP properties, which depend on the type of BAP and ageing processes.](https://doi.org/10.5194/acp-2020-781)

As the number concentration of BAPs only contribute ~0.1% to the total CCN concentration, even under conditions of high N_{BAP}, their role in CCN activation in warm clouds is negligible as they do not lead to any significant change in cloud properties. Since BAPs have usually supermicron sizes, they will act as CCN and even small changes in their chemical composition do not affect their CCN activity. The CCN activation of smaller BAPs such as bacteria fragments or viruses might be influenced by their hygroscopicity (κ_{BAP}) and surface tension (σ_{BAP}). N_{BAP} might be modified by chemical (e.g., nitration, oxidation), physical (e.g., condensation of gases), and biological processes (e.g., formation of metabolic products, biosurfactants). Biosurfactants decrease the surface tension of BAPs (σ_{BAP}) and possibly even of other particles (σ_{other}) to
increase their CCN ability. Even though the CCN activation of BAPs might be of very limited importance for cloud properties, it is more important due to biological aspects as it is a survival strategy of microorganisms to improve their environmental conditions by water uptake, drop formation, and spreading on hydrophobic surfaces to enhance their survival time in the biosphere and atmosphere.

BAPs contribute ~1% to large particles with D > 0.5 µm, which makes them relatively important for the aerosol direct effect. BAPs have a direct cooling effect for most properties explored here whereas they could also change to a direct warming effect for certain species of BAPs. The most sensitive BAP property is the complex refractive index \(m = n + ik\), especially the imaginary part \(k\) which varies by three orders of magnitude among different types of BAPs. Assuming, for example, that all BAPs have the optical properties of *Aspergillus oryzae* fungal spores, the predicted direct aerosol effect changes from a cooling to a warming effect. Our RFE estimates clearly represent an overestimate as we only consider a small particle size range and concentration. Thus, the identified relative changes in RFE due to different BAP types and properties should be considered more representative than the absolute numbers.

The complex refractive index \(m_{\text{BAP}}\) can be modified due to chemical or biological processing. For example, nitrination could lead to an enhancement of the imaginary part (absorptive properties), but the difference of scattering and absorption coefficient induced by nitrination is much smaller compared to the differences caused by the refractive indices of different BAP types. Biological processing such as pigment formation (Pšenčík et al., 2004; Fong et al., 2001) might also lead to \(\Delta m_{\text{BAP}}\) to a significant extent, but we cannot quantify the role of this process in our model framework due to the lack of corresponding data. The second ranked parameter is \(\Delta \theta_{\text{BAP}}\), which also differs among different types of BAP or might change for one BAP type due cell generation or desiccation in the atmosphere. Obviously, the total number of BAP (NBAP) is of importance for all effects discussed here. However, as it has been shown that at many location \(N_{\text{BAP}}/N_{\text{total}}\) is approximately constant, the relative role of BAP likely does not change due to differences in absolute BAP concentration. Hygroscopicity \(\kappa\) might have an effect under high RH conditions. The effect of surface tension \(\sigma\) on direct radiative property is negligible.

The most important role of BAP is to act as IN because \(N_{\text{BAP, IN}} / N_{\text{IN}}\) can reach up to ~100% at \(T > -10 \, ^\circ\text{C}\). Given the high sensitivity of BAPs that initiate freezing, it is clear that not only the total \(N_{\text{BAP}}\) but also the fraction that can freeze needs to be constrained. While this fraction is usually ~100% for pollen, it can be as small as 0.01%-10% for bacteria. As identified in previous sensitivity studies, the surface composition properties, often expressed in terms of a contact angle \(\theta_{\text{BAP}}\), shows the highest importance to IN activity and therefore to the evolution of mixed-phase clouds (Bergeron-Findeisen process). The variability of \(\theta_{\text{BAP}}\) between different types of BAP (4° < \(\theta_{\text{BAP}}\) < 44°) determines the onset temperature of freezing and the temperature interval in which the Bergeron Findeisen process may occur. Even a small change of \(\Delta \theta_{\text{BAP}} \approx 1°\)
as caused by chemical processing on BAP surfaces or pH change might affect the onset of the Bergeron-Findeisen process significantly. Thus, not only various BAP types should be parameterized with different $\theta_{\text{BAP}}$ in models but also $\Delta \theta_{\text{BAP}}$ due to modification by chemical and possibly biological processes.

The trends discussed above are summarized in Figure 1 and show the relative importance of BAPs in the atmosphere, increasing from their roles in CCN activation, to the aerosol direct effect and to mixed-phase cloud evolution. The arrows on the left and on the bottom point to the most sensitive and most important parameters, respectively, which are placed in the upper right corner of the table.

Our study highlights the possible importance of BAP processing as not only chemical and physical processes but also biological ageing processes can modify the chemical composition and physical properties of BAPs. While the former two process types commonly occur on/in many other ambient particles as well (e.g. $\Delta m$ due to nitration of SOA, or $\Delta D$ due to condensation of low volatility material), biological processing is unique to BAPs and currently not comprehensively included and explored in atmospheric models. For example, we suggest that cell generation or the expression of specific proteins might significantly affect BAP’s IN ability. While the role of biosurfactant production ($\Delta \sigma_{\text{BAP}}$) is limited in modulating warm cloud properties and the aerosol direct effect, the biological aspects of this process might be of much larger importance: Enhanced water uptake by BAPs may extend lifetime of the microorganisms by improving their living conditions, i.e. reduce stress due to harsh ambient conditions (e.g. high ionic strength, low pH, desiccation). In addition, their inclusion in clouds as IN or CCN will lead to a more efficient transport and distribution across the atmosphere.

In addition to the few biological processes discussed in our study, additional biological processes (e.g., pigment formation, carotenoid accumulation, formation of metabolic products, biofilm formation) are included in Figure 1 to give a more complete picture of ageing processes of BAP that may affect their radiative properties. Several of our results repeat findings from previous sensitivity studies of aerosol properties on the direct and indirect radiative effects. However, our study should be considered as guidance to future field, lab and model studies to further characterize the role of biological particles in the atmosphere as their emissions, budgets and processing are currently poorly constrained (Khaled et al., 2020) compared to more abundant aerosol types, despite their unique characteristics of living organisms that may affect not only climate but also public health.

**Code and data availability:** Details on the model codes and further model results can be obtained from the corresponding author upon request.

**Author contributions:** MZ and BE designed the model framework. AK, PA, AD contributed by fruitful discussions and commented on the manuscript.
Competing interests: The authors declare that they do not have any competing interests.

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