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Lulu Ning (ninglulu@sust.edu.cn)
Shaanxi University of Science and Technology Xi’an Campus: Shaanxi University of Science and Technology

Huaiqin Ma
Shaanxi University of Science and Technology Xi’an Campus: Shaanxi University of Science and Technology

Qingwen Shi
Shaanxi University of Science and Technology Xi’an Campus: Shaanxi University of Science and Technology

Xuhua Li
Xian Jiaotong University: Xi’an Jiaotong University

Junli Ren
Shaanxi University of Science and Technology Xi’an Campus: Shaanxi University of Science and Technology

Zhijian Li
Shaanxi University of Science and Technology Xi’an Campus: Shaanxi University of Science and Technology

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Effects of Nanocellulose on the Structure of Collagen: Insights from Molecular Dynamics Simulation and Umbrella Sampling

Huaiqin Ma\(^1\), Qingwen Shi\(^1\), Xuhua Li\(^2\), Junli Ren\(^3\), Zhijian Li\(^*\), Lulu Ning\(^*\)

\(^1\) College of Bioresources Chemical and Materials Engineering, Shaanxi University of Science & Technology, Xi’an 710021, China

\(^2\)MOE Key Laboratory for Nonequilibrium Synthesis and Modulation of Condensed Matter, School of Physics, Xi’an Jiaotong University, Xi’an 710049, China

\(^3\)Information Center, Shaanxi University of Science & Technology, Xi’an 710021, China

\(^*\)Corresponding author’s e-mail address: zjli@sust.edu.cn, ninglulu@sust.edu.cn

ABSTRACT: Collagen-nanocellulose composites have been widely used in biomedicine and tissue engineering. However, the detailed mechanism underlying the effects of nanocellulose on the structure of collagen hasn’t been elucidated. As the main component of skin tissue, the conformational disturbance of collagen triggered by nanocellulose may shed light on the biocompatibility of nanocellulose. Therefore, molecular dynamics simulations were carried out to gain insights into the interactions between nanocellulose and collagen. Four different crystal planes of cellulose ((110), (100), (1-10), (010)) have been constructed and the adsorption of collagen onto the four faces has been investigated respectively. It has been found that the structure of collagen remained intact during the binding without chain separation. The intactness of collagen supported the point that the nanocellulose has good biocompatibility. The results derived from umbrella sampling showed that (110) and (1-10) faces exhibit the strongest affinity with collagen, which may be attributed to its hydrophilicity and rather flat surfaces. The hydrophobicity of (100) facet and roughness of (010) facet diminished the affinity with collagen. The occupancy of hydrogen bonds was low and hydrogen bonding interactions fail to make significant contributions to the binding of nanocellulose and collagen. These findings provided insights into the interactions between cellulose and collagen at an atomic level, which may guide the design and fabrication of collagen-nanocellulose composites. Furthermore, the biocompatibility of nanocellulose validated in the study may help promote the biological application of nanocellulose involved composites.

Keywords: nanocellulose, collagen, molecular dynamics simulation, umbrella sampling
1. Introduction

As a valuable new type of green biological nanomaterial, nanocellulose has the advantages of a fine nanostructure, good mechanical strength, and low thermal expansion coefficient, as well as recoverability and sustainability (Fernandes and Madhuranthakam 2020; Zhang et al. 2019). Moreover, nanocellulose in the form of nanoparticles, tablets, gas-liquid gel system, and fiber membrane system is widely used in the field of energy storage (Chen et al. 2018), wastewater treatment (Tang et al. 2019), 3D bioprinting (Dorishetty et al. 2020), biosensor (Golmohammadi et al. 2017), drug delivery (Salimi et al. 2019) and other biological fields. Compared with graphene (Luan, Huynh and Zhou 2016), graphene oxide (He et al. 2019), molybdenum disulfide (Gu et al. 2016; Gu et al. 2017), carbon nanotube (El-Sayed et al. 2016), and other 2D nanomaterials, nanocellulose as a carrier material has a lower nanotoxicity, which makes it more widely available. Although nanocellulose and its derivatives have been widely concerned in medicine and biological tissue, the long-term retention of nanocellulose in the human body makes its toxicity study very important due to the lack of cellulose-degrading enzymes in the human body.

Collagen can also be a candidate for biomaterials such as tissue-engineered scaffolds and wound dressings (Lee et al. 2019; Ge et al. 2018; Sorushanova et al. 2019). However, the application of pure collagen materials is limited due to their low water resistance, fast biodegradation perishability, and poor thermal stability (Ge et al. 2018). While cellulose and collagen nanocomposite materials overcome the weaknesses of pure collagen materials. Cellulose and its derivatives can be widely used to strengthen various polymer matrix materials due to their high specific surface area, high crystallinity, low density, and high elastic modulus (Manhas et al. 2015; Salimi et al. 2019; Liu et al. 2018; Li et al. 2017). Researchers have proved that collagen/nanocellulose composite has good properties and stability better than pure collagen. Animal experimental studies (Liu et al. 2020b; Liu et al. 2020a) (Collagen/cellulose nanofiber hydrogel scaffold: physical, mechanical and cell biocompatibility properties; A 3D porous microsphere with multistage structure and component based on bacterial cellulose and collagen for bone tissue engineering (Zhang et al. 2020); demonstrated that collagen and nanocellulose composite is a promising material for wound dressings and tissue engineering scaffolds.

The advantages and disadvantages of nanocellulose and collagen can effectively complement each other to form a more potential nanocomposite material (Cudjoe et al. 2017), which makes their
composites have a broader application prospect. However, the interactions between nanocellulose and collagen, which are significantly related to the strength of composites, are still obscure. Furthermore, in vivo and in vitro experiments have shown that nanocellulose and its derivatives have adverse effects on intestinal microorganisms (DeLoid et al. 2019), liver cells (Otuechere et al. 2020), and lung cells (Sai and Fujita 2020). In addition, nanocellulose biological dressings and tissue-engineered materials, as the main components of human tissues, will directly interact with collagen when they contact the human body. Therefore, it is necessary to study the toxicity of nanocellulose.

In this study, molecular dynamics simulations were carried out to study the interactions between collagen and nanocellulose. Nanocellulose materials are characterized by a high degree of crystallinity thus it is feasible to employ cellulose crystal to model the properties of nanocellulose. Native crystalline cellulose contains a mixture of faces, which complicates the direct assignment of cellulose-collagen interactions to a specific face by experimental methods such as NMR spectroscopy. Molecular dynamics simulations provide a good solution to bypass this limitation. A single crystal face in nanoscale can be constructed by molecular modeling. Molecular dynamics simulations have been carried out by Crowley et al. to study the interactions between cellulose and lignin and gain insights into quantitative relationships between different cellulose faces and specific lignin chemistries (Vermaas, Crowley and Beckham 2019). Moreover, molecular dynamics simulations have also been used to probe the interactions between water molecules and cellulose, which shed light on the wetting mechanisms of cellulose (Malaspina and Faraudo 2019); Cellulose nanocrystals produced using recyclable sulfuric acid as hydrolysis media and their wetting molecular dynamics simulation; Molecular simulation of surface reorganization and wetting in crystalline cellulose I and II, Wetting the (110) and (100) Surfaces of Iβ Cellulose Studied by Molecular Dynamics. Furthermore, many the interaction between two-dimensional nanomaterials and biomacromolecules have also been investigated by molecular dynamics simulations. Zhou et al., adopted molecular dynamics simulation (MD) to conduct thorough research on the interfacial properties of two-dimensional nanomaterials and found that graphene (Luan et al. 2016), graphene oxide (He et al. 2019; Mathesh et al. 2016), defective graphene (Gu et al. 2019) and other carbon nanomaterials can induce protein and nucleic acid denaturation, which provided theoretical support for the potential nanotoxicity of nanomaterials when used in biological systems.

In this work, Iβ-cellulose, which is one of the main components of higher plants, was selected to
model nanocellulose. Collagen type I was employed as the model collagen, which is the most abundant and widely distributed natural structural protein in the human body (Lin and Liu 2006). The adsorption of collagen on different crystal planes of cellulose was simulated by molecular dynamics. Then the structural changes and specific interactions were characterized in detail. This study investigated the interactions between nanocellulose and collagen at a molecular scale and evaluated the structural changes of collagen, which revealed the possible biological effects of nanocellulose and provided theoretical guidance for the design of nanocellulose-collagen complex at the same time.

2. Methods

The initial structure of collagen is obtained by extracting three chains from the crystal structure (Berisio et al. 2009) (PDB code 1K6F) as shown in Fig. 1(B) All nanocellulose crystal faces of different sizes (Fig. 1A) were constructed using Cellulose-Builder (Gomes and Skaf 2012). The thickness of (100), (110), (1-10) and (010) varied to weaken the effects of surface tension. Collagen was placed above the surface of the nanocellulose crystal at different angles (0°, 30°and 45°) in VMD software (Humphrey, Dalke and Schulten 1996) to form the initial coordinates of the simulation system and the minimum distances between cellulose faces and collagens were ranging from 0.8 nm to 1 nm (Fig. 2A). As shown in Fig. 2(A), three parallel simulations of each system were carried out for 500 ns. The composite system was solvated in a cubic box with a TIP3P water model (Mark and Nilsson 2001; Jorgensen et al. 1983) and modeled by a CHARMM36 force field (Lee et al. 2014; Boonstra, Onck and van der Giessen 2016).

Fig. 1. (A) The crystalline plane of cellulose Iβ. (B) Collagen in initial configuration (PDB ID: 1K6F).

All MD simulations were performed in GROMACS-5.1 (Berendsen, Spoel and Drenen 1995; Van Der Spoel et al. 2005) package All the systems were equilibrated carefully in the beginning of simulation. The energy minimization process was carried out with 1000 cycles of steepest descent and 1,000 cycles
of conjugate gradient minimization. Then, equilibration runs were performed for 5 ns in the NVT ensemble and 5 ns in the NPT ensemble with the heavy atoms of protein and cellulose fixed. Finally, 500 ns production runs were simulated in the NPT ensemble with the restriction of the protein released. The long-range electrostatic interactions were treated by the particle mesh Ewald (PME) method (Petersen 1995), while the short-range van der Waals interactions were calculated with a cutoff distance of 1.0 nm. All covalent bonds containing hydrogen atoms were constrained by the LINCS algorithm (Hess et al. 2008). The V-rescale thermostat (Berendsen et al. 1984) was used to heat the system to 300 K and the Parrinello-Rahman Pressure coupling (Parrinello and Rahman 1981; Nosé and Klein 2006) kept the system pressure at 1 bar. The integration step size of the simulation process is 2 fs. Periodic boundary conditions were applied in all directions with glycosidic bonds formed between mirror images.

The potential of mean force (Roux 1995) (PMF) obtained by pulling simulation and umbrella sampling (Hub 2015) was used to calculate the binding free energy of the system. The cellulose surface was used as a reference point and a harmonic potential was applied to the collagen as a pulling point. The last frame of the MD simulations was selected as the initial conformation, 300 ps umbrella traction was provided for collagen along the z-axis to increase the center of mass (COM) distance between collagen and cellulose. The spring constant used was 2000 kJ mol$^{-1}$ nm$^2$ and the pull rate was 0.01 nm/ps. More than 13 umbrella sampling windows were selected according to the interval size of COM values. 1ns of simulations in NPT was performed on each sample, then 10 ns of MD process was carried out. Finally, weighted histogram analysis (Hub 2015) (WHAM) was used to calculate PMF.

The relevant modules in GROMACS were used to calculate the backbone root mean square deviation (RMSD) and backbone root mean square fluctuation (RMSF) of proteins during the whole simulation process. G_hbond was used to calculate the number of hydrogen bonds the distance of 0.30 nm and angle 30° as criterial.

3. Results and discussions

3.1 Structural of collagen remained intact on nanocellulose

In the initial simulation system, the distance between protein and cellulose crystal faces was controlled between 0.8-0.1nm as shown in Fig. 2A and three groups of different model analyses were conducted for each crystal face. Although the hydrophilicity and hydrophobicity of the selected cellulose
crystal faces were different, the adsorption behaviors of collagen on different crystal faces were similar. The protein slowly contacted the cellulose surface. As shown in Fig. 2(B-E), no significant structural changes were observed except that the overall structure of collagen was slightly bent at the end of the simulation.

Fig. 2. (A) The initial system configuration of (100) planes. The water box is rendered with a cyan surface. (B-E) Snapshots of four crystal planes at 500ns. The protein shows green (0°), purple (30°), and blue (45°) to indicate three tracks, and the cellulose crystal plane is shown in sphere (carbon, gray; oxygen, red), hydrogen is not shown for clarity.

The calculation of the root-mean-square deviation of the protein was carried out to quantitatively measure the change of collagen structure during the simulation. RMSD values of collagen on different nanocellulose crystal faces are kept between 0.2 nm-0.3 nm as shown in Fig. 3(A-D). The profiles of RMSD exhibit very small fluctuation during the whole process. RMSD values in this range indicated that the protein structure did not change significantly during the adsorption process. Furthermore, RMSF was calculated to evaluate the free movement degree of each residue in collagen molecules. As shown in Fig. 3(E-H), the profiles of RMSF has good accordance with each other regardless of the properties of different cellulose faces. Each chain of the collagen model is composed of 29 residues and the peak of the line represents the end of each chain, which indicates that the two ends of the polypeptide chain are more flexible and the structure of other residues located in the middle of collagen is stable. In conclusion, the overall structure of collagen was not damaged during the process of adsorption on different crystal faces of cellulose. It is widely recognized that the change of protein structure induces the loss of its biological function (Feng et al. 2017), thus it can be inferred that nanocellulose has good biocompatibility.
3.2 Evolution of the secondary structure of collagen

The structure of proteins plays an important role in their activity and biological effects. To analyze the structural changes of collagen in the process of adsorption in more detail, the Ramachandran plot (Hooft, Sander and Vriend 1997) was used to characterize the changes in the secondary structure of proteins. Collagen is a coil, but one with distinct tertiary and quaternary structures: three separate polypeptides, called α chains are supertwisted about each other. The superhelical twisting is right-handed in collagen, opposite in sense to the left-handed helix of the α chains. Thus, it is ambiguous to evaluate the secondary structure of collagen by designating the structure helix, sheet, or coil. Therefore, Ramachandran plots were employed. The collagen helix is a unique secondary structure with Phi=-51° and Psi=+153°, which is quite distinct from the α helix (Köppen, Ohler and Langel 2007). As shown in Fig. 4, Ramachandran plots of the last frames extracted from MD simulations display similar characteristics and most of Phi and Psi values are confined within the region corresponding to the structure of collagen. Random coil structures have also been observed with Phi and Psi values appearing around (-90, -180). It was found by comparison that the Phi and Psi angles of collagen not significantly deviated (Fig. 4A-D) from the specified collagen conformation values during the adsorption of collagen in different crystal planes of cellulose. Therefore, it is indicated that all models exhibit partial structural interruption but are not damaged during adsorption.
Fig. 4. The Ramachandran plots of collagens adsorbed on the planes (100) (A), (110) (B), (1-10) (C), and (010) (D).

To further investigate the effects of nanocellulose on the global structure of collagen, principal component analysis contour maps were constructed, which is also an effective method to analyze protein structure. Fig. 5 displays the free energy contour map of collagen after PCA in 12 simulated trajectories. As shown in Fig 5, the free energy contour values of each model are located in a similar region with only one global minimum, which indicates that the effects of different crystal faces on collagen structures are too little to induce obvious changes. All the global minima of the 12 MD simulations are restricted within narrow ranges with RMSD about 0.23 nm and Rg about 0.42 nm, which further indicates the intactness of collagen on the surface of nanocellulose. The conformational space of collagen on the surface of nanocellulose provided information about the structural state of collagen in nanocellulose-collagen composites. Moreover, the study also partially evaluated the biological effects of nanocellulose when it is applied in the human body as a biomedical material. It exerted limited influence on the structure of collagen, which further validated the biocompatibility of nanocellulose.
3.3 Interaction between collagen and nanocelluloses during adsorption

To understand the adsorption of collagen onto nanocellulose, umbrella sampling simulations were performed for all the four faces. Prior to the umbrella sampling simulation, the centers of all the cellulose models move to original points. Reaction coordinates were corrected by deducting half of its thickness in order to eliminate the effects of different thicknesses of cellulose slab and better illuminate the properties of different surface morphologies. As shown in Fig. 6(A), the lowest PMF values are found near the layer surface for all the systems, indicating that collagen tends to bind to cellulose. PMF shows that free energies are close for faces (110) and (1-10) with values of -15.5 kcal/mol and -14.8 kcal/mol respectively, which imply that almost the same adsorption strength of collagen onto the (110) and (1-10) surface. Free energies of (100) (-9.5 kcal/mol) are a little higher than those of (110) and (1-10) faces, indicating that the affinity between (100) layer and collagen are slightly weaker. In contrast with PMF profiles of (110), (1-10), and (100), the binding energies of (010) are much lower (-5.2 kcal/mol). These free energies derived from umbrella samplings indicate that collagen tends to migrate toward all the faces.
of cellulose. In particular, (110) and (1-10) faces displaying the strongest affinity to collagen. (100) face, which is more hydrophobic, displays weaker adsorptive capability with collagen. Due to the structural anisotropy of cellulose, Iβ(110) and Iβ(1-10) faces are hydrophilic while Iβ(100) face is hydrophobic. The chain of collagen is composed of repeating tripeptide sequence Gly-Pro-Pro, which are all polar amino acids. Polar interactions between hydrophilic cellulose faces and collagen enhance their affinity. Though (010) face is hydrophilic, the interaction between these faces and collagen is the weakest among the four systems, which seems quite counterintuitive. The origin of this behavior is attributed to the topography of (010) faces, which hinders the binding of collagen with half of the hydroxyl grouping shielded. The steric hindrance weakens the interactions between collagen and (010) face, which is consistent with the results of the analysis of contact number. (110) and (1-10) are more hydrophilic than (100) but less rough than (010) face (Fig. 7E-H), which may result in a stronger affinity with collagen.

Fig. 6. (A) Potential of mean force plots describing the binding of collagen and cellulose. The radial distribution function of oxygen in exposed hydroxyl groups on water and cellulose surfaces, with cellulose surfaces represented by (B) (100), (C) (110), (D) (1-10) and (E) (010) respectively. To further validate the deduction, radial distributions of water molecules were calculated to evaluate the hydrophilicity of the four faces. As shown in Fig. 6(B-E), their first peak positions occur at the same position 3.8 Å, respectively. The height of the first g (r) peak belonging to (100) face is about 0.4, which is significantly lower than that of (110), (1-10) and (010) faces with the heights of the first peak all about 0.6. Thus, the heights of the first peaks describe a distinguishable difference in hydrophilicity among the four faces and the surface hydration of (100) face is weaker than the other three faces. To evaluate the effects of surface morphology on the interaction between collagen and cellulose, the contact number of
heavy atoms was calculated with 0.5 nm as the threshold. In general, the loading of collagen on the
cellulose surface was fast with molecules of collagen adsorbed on cellulose within 200 ns. Based on the
heavy atom contact numbers between collagen and cellulose, it has been deduced that there is an obvious
correlation between surface roughness and contact numbers. As shown in Figure 7, (100) face displays
the largest contact number with the smoothest surface, while, (010) face exhibits the least contact number
with the greatest surface roughness. (110) and (1-10) faces are in between with the contact numbers larger
than that of (100) face but less than that of (010) face. As the fundamental part of molecular interactions,
the block of direct contact between collagen and (010) face impaired their affinity. The combinations of
mediocre hydrophilicity and smoothness made (110) and (1-10) faces stand out from the other two crystal
facets in the adsorption process.

Fig. 7. Change of heavy atomic contact number of collagen and (100) (A), (110) (B), (1-10) (C), (010)
(B) faces over time. (E-H) The last frame trajectory of the model when the collagen was 0° on the four
crystal planes. Collagen is blue, cellulose overall structure is gray.

Cellulose molecules contain a large number of free hydroxyl groups, which might be involved in
hydrogen bonding interactions. Therefore, the average occupancy of hydrogen bonds between the
collagen and cellulose in the four systems was calculated respectively. As shown in Fig. 8, the values of
hydrogen bonding occupancies are all lower than 12%, which indicates that hydrogen bonding interaction
is not the predominant force driving the binding of collagen and cellulose. (010) face of cellulose exhibits
the highest tendency to form hydrogen bonds with collagen, which is consistent with the outward
orientation of its surface hydroxyl groups. The hydrophobic (100) facet is less inclined to be involved in
hydrogen bonding interactions with hydroxyl groups mainly forming interchain hydrogen bonds. (110)
and (1-10) facets have a median performance in hydrogen bonding interactions. Collagen adsorption may
be mainly driven by the dispersion interaction between the collagen and cellulose surface.

Fig. 8. Hydrogen bond occupancy between collagen and cellulose crystalline faces.

4. Conclusion

In this study, molecular dynamics simulations were carried out to investigate the adsorption behavior of collagen on the ideal nanocelluloses surface. It has been observed that the structural integrity of collagen has been maintained in the process of adsorption, which may shed light on the biocompatibility of cellulose. (110) and (1-10) crystal faces exhibited the strongest affinity with collagen, which was attributed to the combination of hydrophilicity and roughness. Hydrogen bonding interactions are not frequent and are not the predominant force driving the binding of collagen and nanocellulose. This study provides theoretical guidance for the design and fabrication of collagen-nanocellulose composites. Furthermore, the intactness of collagen structure supported the viewpoint that nanocellulose is quite biocompatible.

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