1. Introduction

With the development of space science and technology, geosynchronous orbit satellites are being developed with lightweight, long life and high reliability.\(^1,2\) Thermal control coating is a main passive thermal control means to ensure the temperature balance of the satellite,\(^3,4\) guaranteeing the highly reliable operation of the satellite. Therefore, it is desired to develop a novel thermal control coating with lightweight and high space stability to improve the service life of satellites.

At present, thermal control coatings are mainly divided into organic coatings and inorganic coatings according to composition.\(^5\) In particular, inorganic thermal control coatings have a low radiation-absorption ratio and excellent stability in the space environment, which is preferred for spacecraft cooling surface at present.\(^7,8\) Because the thermal control coating is located on the outer surface of the spacecraft, it will be directly exposed to the harsh space environments, such as high vacuum, charged particles, ultraviolet radiation, atomic oxygen and so on.\(^9,10\) Among the harsh space environments, it is worth noting that the vacuum-ultraviolet radiation will cause the serious degradation of thermal control performance and affect its function because high-energy UV photons can cause “color center” generation on the inorganic thermal control coating surface.\(^11\) Thus, it is important to improve vacuum-ultraviolet resistance performance of inorganic thermal control coating.

Zinc oxide (ZnO), as excellent ultraviolet resistant material,\(^14\) which is widely chosen to prepare the inorganic thermal control coating of spacecraft because of the matching between band gap width and ultraviolet wavelength, chemical stability at high temperature, and permanent stability under UV exposure. Meanwhile, studies have found that rare earth modified ZnO-based micro-arc oxidation coating can effectively reduce the UV absorption capacity,\(^15\) and Y\(_2\)O\(_3\) exhibits unique electrical, magnetic and chemical properties due to their unique electronic structure, which can inhibit the crystal phase transition of metal oxides,\(^16,17\) and thus improve the stability of thermal control coating. Accordingly, introducing ZnO and Y\(_2\)O\(_3\) into the inorganic thermal control coating can drastically improve its vacuum-ultraviolet resistance performance.

Herein, in this work, an inorganic Al\(_2\)O\(_3\)–ZnO–Y\(_2\)O\(_3\) thermal control coating was in situ fabricated on the 6061 aluminum alloy surface by plasma electrolytic oxidation (PEO) technology. The vacuum-ultraviolet resistance performance of Al\(_2\)O\(_3\)–ZnO–Y\(_2\)O\(_3\) thermal control coating was investigated by testing and analyzing its thermal control performance, mass loss ratio, surface microstructure and phase structure before and after vacuum-ultraviolet irradiation 2000 ESH. Furthermore, the life prediction model of the Al\(_2\)O\(_3\)–ZnO–Y\(_2\)O\(_3\) thermal control coating shows that its \(\Delta \alpha\) value first slightly increases and then tends to be stable with the increase of ultraviolet irradiation time from 0 ESH to 25 000 ESH, and the maximum variation of \(\Delta \alpha\) is about 0.0536. This work provides a material basis and technical support for the thermal control system of spacecraft with long life and high reliability.
prediction model of $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating under vacuum-ultraviolet irradiation is preliminarily established. This work provides a material basis and technical support for the thermal control system of spacecraft with long life and high reliability.

2. Experimental

2.1 Materials

Sodium polyphosphate ($\text{Na}_2\text{H}_2\text{P}_2\text{O}_7$), sodium metasilicate ($\text{Na}_2\text{SiO}_3$), zinc dihydrogen phosphate ($\text{Zn}[\text{H}_2\text{PO}_4]_2$) and ethylenediamine tetraacetic acid disodium salt ($\text{Na}_2\text{EDTA}$) were provided by Tianjin Chemical Reagent Company, China. Yttrium nitrate ($\text{Y(NO}_3)_3$) and potassium hydroxide ($\text{KOH}$) were purchased from Aladdin Reagent Company. Other chemical reagents were supplied by Harbin local reagent company. All reagents were analytical grade and used without further purification.

2.2 Preparation of $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating

$\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating was fabricated on the 6061 aluminum alloy surface with a dimension of 4 cm × 4 cm × 0.2 cm by plasma electrolytic oxidation (PEO) technology. Prior to the PEO treatment, aluminum alloy surface was polished by using 150, 400 and 800 mesh abrasive paper, respectively, and ultrasonically degreased in acetone and then cleaned with the distilled water. The 20 kW bipolar pulsed power supply was employed to carry out the PEO reaction of the samples in the electrobath of the stainless steel serving as the cathode. The power parameters were fixed at the pulse frequency of 500 Hz, the current density of 10 A dm$^{-2}$, the duty ratio of 30% and oxidation time of 15 min. The electrolyte solution consists of 30 g L$^{-1}$ $\text{Na}_2\text{H}_2\text{P}_2\text{O}_7$, 20 g L$^{-1}$ $\text{Na}_2\text{SiO}_3$, 10 g L$^{-1}$ $\text{Zn}[\text{H}_2\text{PO}_4]_2$, 10 g L$^{-1}$ $\text{Na}_2\text{EDTA}$, 8 g L$^{-1}$ $\text{KOH}$ and 6 g L$^{-1}$ $\text{Y(NO}_3)_3$·6H$_2$O. After the PEO process, the obtained samples were rinsed with water and then dried in the air.

2.3 Simulation experiment of vacuum-ultraviolet irradiation

The vacuum-ultraviolet irradiation simulation experiment in space of $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating on the aluminum alloy surface was operated in the near ultraviolet irradiation area under the power of 1000 W mercury xenon lamp with the infrared and visible light filter, and the test related parameters mainly include the vacuum degree below 1 × 10$^{-5}$ Pa, irradiance of 4 SC, sample temperature of 30 °C and heat sink temperature below −35 °C. The vacuum-ultraviolet irradiation amount is 500–2000 ESH (equivalent solar hour: the total irradiation amount corresponding to the sample irradiated for 1 hour with a solar constant), and the solar absorptivity ($\alpha_s$) and emissivity ($\varepsilon$) of the selected sample before and after vacuum-ultraviolet irradiation were measured to evaluate the vacuum-ultraviolet resistance performance through an in situ measurement system in the equipment.

2.4 Characterization

The coating thickness of the $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating was measured through the coating thickness gauge (TT260, Beijing Times Peak Technology Co., Ltd). The microstructure of $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating was observed by scanning electron microscope (SEM, Hitachi, S-570). 3D surface structure was measured by atomic force microscope (AFM, CSPM 5500). The crystal structure of $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating was characterized by X-ray diffractometer (XRD, Rigaku, Dmax-3B) with a Cu Kα radiation ($\lambda = 1.54060$ Å). The emissivity ($\varepsilon$) at 2–16 μm was examined by an infrared reflectometer (Gier-Dunkle DB-100) and the solar absorptivity ($\alpha_s$) in the 200–2500 nm was measured by a UV-VIS-NIR spectrophotometer (PerkinElmer Lambda 950) with an integrating sphere.

3. Result and discussion

3.1 Thermal control performance

As we all know, the coating thickness plays an important role in its thermal control performance. Under the condition of satisfying the requirements of thermal control performance of spacecraft, weight reduction of spacecraft still needs to be considered. Therefore, in this work, the thickness of the as-prepared $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating is controlled at about 30 μm. According to the test method B-Cross-Cut tape test of ASTM-D3359-17 standard, the classification of adhesion test results for the obtained $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating is 5B. Fig. 1 is the emissivity and absorptivity variation of $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating at different vacuum-ultraviolet irradiation time. As shown in Fig. 1, it can be seen that the initial emissivity and absorptivity of the $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating are 0.859 and 0.405, respectively, which meet the application requirements. With the increase of vacuum-ultraviolet irradiation time from 0 ESH to 2000 ESH, its emissivity cannot decrease but slightly increase from 0.405 to 0.435 (Fig. 1b), indicating that the vacuum-ultraviolet irradiation has no adverse effect on the emissivity of $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating. On contrast, although its absorptivity also presents a slight increase from 0.405 to 0.435 (Fig. 1a) and the variation is very little, which can still meet the application requirements. The above results show that vacuum-ultraviolet irradiation has little effect on thermal control performance of $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating.

3.2 Mass loss ratio

The mass loss ratio variation of $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating on aluminum alloy surface under different vacuum-ultraviolet irradiation time is shown in Fig. 2. It can be seen that the mass loss ratio of $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal control coating shows an increase trend within the vacuum-ultraviolet irradiation time of 1000 ESH, which is attributed to the precipitation of adsorbed impurities and degradation product. Continuing to increase the vacuum-ultraviolet irradiation time from 1000 ESH to 2000 ESH, its mass loss ratio basically maintains a stable value of about 0.50 mg cm$^{-2}$, which can meet the requirements of mass loss for spacecraft thermal control coating, indicating that $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal coating can meet the requirements of mass loss for spacecraft thermal control coating, indicating that $\text{Al}_2\text{O}_3$–ZnO–$\text{Y}_2\text{O}_3$ thermal coating can meet the requirements of mass loss for spacecraft thermal control coating.
control coating has a good stability in vacuum-ultraviolet environment.

3.3 Surface morphology analysis

In order to investigate the effect of vacuum-ultraviolet irradiation on surface morphology of Al₂O₃–ZnO–Y₂O₃ thermal control coating, SEM and AFM were used to observe the surface morphology variation of Al₂O₃–ZnO–Y₂O₃ thermal control coating before and after vacuum-ultraviolet irradiation 2000 ESH, as shown in Fig. 3. It can be seen that the surface morphology of Al₂O₃–ZnO–Y₂O₃ thermal control coating cannot be changed after vacuum-ultraviolet irradiation 2000 ESH, indicating that the surface morphology of thermal control coating has a good stability under vacuum-ultraviolet irradiation.

3.4 Phase structure analysis

The phase structure variation of Al₂O₃–ZnO–Y₂O₃ thermal control coating can be determined by XRD measurement. Fig. 4 is XRD patterns of Al₂O₃–ZnO–Y₂O₃ thermal control coating before and after vacuum-ultraviolet irradiation 2000 ESH. It can be seen that the main diffraction peaks location of Al₂O₃–ZnO–Y₂O₃ thermal control coating before and after vacuum-ultraviolet irradiation 2000 ESH cannot be changed, indicating that the phase structure of thermal control coating also has a good stability under vacuum-ultraviolet irradiation.

3.5 Vacuum-ultraviolet resistance mechanism

From the analysis of atomic structure, when the ultraviolet photon irradiates the Al₂O₃–ZnO–Y₂O₃ thermal control coating, the electrons around the lattice atoms at the irradiation will generate the deviation. When the deviation is accumulated to a certain extent, the spectral absorption effect will be generated, causing the increase of the absorption ratio (aₛ). The number of lattices without color center, the number of existing color center lattices and ultraviolet irradiation were positively correlated with the negative exponent of the product of color center formation probability and incident intensity. The material is considered to be an ideal material. The lattice number of color center generation in each molecular layer is N₀, and the probability of color center generation in the inner molecular layer is μ. Therefore, the relation between color center concentration (Cₜ) and irradiation time (T), color center generation probability is shown in formula (1).

\[
C_\text{d} = \frac{N(X, T)}{N_0} = 1 - \exp[ -T \mu \exp(-BX) ]
\]

(1)

where X is penetration depth of ultraviolet irradiation photons, and Ω is the wavelength absorption coefficient of the thermal control coating in the ultraviolet band region. According to the above formula (1), there are the two limiting conditions with the change of irradiation time: (1) when T = 0, N(X, T) = 0, Cₜ = 0, αₛ(T) is the initial value αₛ; (2) when T \rightarrow \infty , N(X, T) \rightarrow N₀, Cₜ \rightarrow 1, αₛ(T) tends to the limit value αₛ. Based on above two limit conditions, formula (1) was integrated to obtain the formula (2).
$\alpha_s(T) = \alpha_0 \int_0^D \frac{N_0 - N(X, T)}{N_0} \, dX + \alpha_U \int_0^D \frac{N(X, T)}{N_0} \, dX \quad (2)$

where $D$ is the total thickness of the thermal control coating. Since ultraviolet irradiation has a threshold of penetration depth $d_U$, the formula (2) can be transformed into the following formula (3), and thus a mathematical model of solar absorption degradation of thermal control coatings under ultraviolet irradiation can be obtained.

$$\Delta \alpha_s(T) = D^{-1} (\alpha_U - \alpha_0) \int_0^{d_U} \{1 - \exp[-T \mu_0 \exp(-BX)]\} \, dX \quad (3)$$

According to the formula (3), it can be seen that $\Delta \alpha_s(T)$ is related to ultraviolet extinction coefficient of thermal control coating when the external environment, ultraviolet irradiation intensity and time are fixed. Therefore, the material with a large extinction coefficient in the ultraviolet band is conducive to improving the ultraviolet irradiation resistance of the material.

Table 1  The absorptivity variation ($\Delta \alpha_s$) of $\text{Al}_2\text{O}_3$–$\text{ZnO}$–$\text{Y}_2\text{O}_3$ thermal control coating

| Ultraviolet irradiation time (ESH) | After ultraviolet irradiation $\alpha_s$ | Variation $\Delta \alpha_s$ |
|-----------------------------------|----------------------------------------|---------------------------|
| 0                                 | 0.405                                  | 0                         |
| 500                               | 0.414                                  | +0.009                    |
| 1000                              | 0.423                                  | +0.018                    |
| 1500                              | 0.427                                  | +0.022                    |
| 2000                              | 0.434                                  | +0.029                    |
| 3000                              | 0.441                                  | +0.036                    |
| 4000                              | 0.446                                  | +0.041                    |
| 5000                              | 0.451                                  | +0.046                    |
Fig. 5 The prediction $\Delta a_t$ degradation curve of $\text{Al}_2\text{O}_3$–$\text{ZnO}$–$\text{Y}_2\text{O}_3$ thermal control coating under vacuum-ultraviolet irradiation.

Al$_2$O$_3$–ZnO–Y$_2$O$_3$ thermal control coating has a good resistance performance to vacuum-ultraviolet radiation.

4. Conclusion

The inorganic Al$_2$O$_3$–ZnO–Y$_2$O$_3$ thermal control coating was in situ fabricated on the 6061 aluminum alloy surface by PEO technology. The vacuum-ultraviolet resistance performance of Al$_2$O$_3$–ZnO–Y$_2$O$_3$ thermal control coating was investigated. The results show that Al$_2$O$_3$–ZnO–Y$_2$O$_3$ thermal control coating has good resistance performance to vacuum-ultraviolet irradiation, which is mainly because the large extinction coefficients of ZnO and Y$_2$O$_3$ materials in the ultraviolet band is conducive to improve the vacuum-ultraviolet resistance performance. Furthermore, the life prediction model of Al$_2$O$_3$–ZnO–Y$_2$O$_3$ thermal control coating shows that its $\Delta a_t$ is only 0.536 under vacuum-ultraviolet irradiation time less than 7000 ESH, and then its $\Delta a_t$ tends to be stable value with the increase of ultraviolet irradiation time from 7000 ESH to 25 000 ESH. This work provides a material basis and technical support for the thermal control system of spacecraft with long life and high reliability.

Author contributions

Tao Li and Hongjun Kang are roles for conceptualization, investigation, writing-original draft. Songtao Lu is role for investigation and validation. Wei Qin and Xiaohong Wu are role for resources and data curation.

Conflicts of interest

There are no conflicts to declare.

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