A tutorial on oxidative stress and redox signaling with application to exercise and sedentariness

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Abstract
Oxidative stress has been shown to play a role in the etiology of several chronic diseases, including cardiovascular disease, diabetes mellitus, and cancer. Free radicals and, most prominently, the superoxide radical, result from oxidative metabolism and several enzyme-catalyzed reactions, and endogenous cellular antioxidants dismutate many reactive oxygen species (ROS). Under certain conditions, ROS production can outpace dismutation (e.g., long-term sedentariness and positive energy balance) and the result is oxidative stress, with proteins, lipids, and DNA the most common targets of radicals. However, the molecules that contribute to oxidative stress also appear to participate in vital cell signaling activity that supports health and stimulates favorable adaptations to exercise training, such that inhibiting ROS formation prevents common adaptations to training. Furthermore, researchers have recently suggested that some proteins are not as readily formed when the redox state of the cell is insufficiently oxidative. Exercise training appears to optimize the redox environment by dramatically enhancing the capacity of the cell to neutralize ROS while regularly creating oxidative environments in which membrane and secretory proteins can be synthesized. The role that exercise plays in enhancing management of ROS likely explains many of the associated health benefits.

Key points
- Reactive oxygen species (ROS) in excess are toxic and have been implicated in the development of aging and chronic disease.
- ROS have also been shown to enhance acute muscular activity, and to be required signaling molecules for favorable adaptations to exercise training.
- Long-term sedentariness is characterized by chronically elevated basal ROS production and reduced antioxidant capacity.
- The oxidative environment induced by exercise stimulates antioxidant capacity and may enhance synthesis of specific proteins.

Introduction
Oxidative stress has been implicated in the etiology of a number of chronic diseases including cardiovascular disease, diabetes mellitus, and cancer. It has been defined as an imbalance between oxidants and antioxidants in favor of oxidants, leading to the disruption of redox signaling and control and/or molecular damage [1]. Recent studies have shed a great deal of light on factors that influence oxidative stress. Exercise training is among the factors known to provide protection against oxidative stress, and understanding how exercise training improves oxidative status may be helpful to exercise professionals. The purposes of this review are to (1) explain what oxidative stress is, and how and where it occurs; (2) summarize recent studies that suggest that some minimal level of oxidative molecules are essential for cell signaling and stimulating beneficial adaptations to training; (3) explain how oxidative agents, in excess, are damaging; and (4) explain the effects of acute exercise and chronic exercise training on oxidative stress. As the major focus of this paper is exercise, we primarily consider these concepts within muscle fibers. However, exercise appears to stimulate reductions in signs of oxidative stress systemically (e.g., in brain and liver tissue), and those influences are also addressed briefly.

Oxidation, reduction, and radicals
Oxidation occurs when a molecule loses an electron. When one molecule loses an electron, another will acquire it, and the molecule that gains an electron is said to be reduced. Oxidation and reduction, then, occur
together, in that whenever a molecule is oxidized, another is reduced. The fact that they occur together has led to the development of the term ‘redox’ being used in the context of this class of chemical reactions.

While there are countless examples of redox reactions in biochemistry, the reactions that are most closely related to oxidative stress involve molecules that are especially strong oxidizing agents, known collectively as free radicals, and include reactive oxygen species (ROS) and reactive nitrogen species (RNS) (the predominance of RNS is often referred to as nitrosative stress). These molecules are especially strong oxidizing entities because they have an unpaired electron in their outer shells. Oxygen and nitrogen are atoms that display a high level of electronegativity, owing to the size of the atoms and the number of protons in their nuclei. The result is that these atoms have a strong attraction for electrons when there are no unpaired electrons in their outer shells, and that attraction is much stronger when there is an unpaired electron.

One commonly produced radical, and one that is the basis of other oxidative agents, is the superoxide radical—molecular dioxide ($O_2^-$)—that has undergone a one-electron reduction. Superoxide reacting with other molecules can result in other radicals or oxidizing agents being produced. For example, much of the superoxide that is produced in mitochondria is dismutated or partially neutralized by an enzyme known as manganese superoxide dismutase (MnSOD), a protein that converts superoxide to hydrogen peroxide ($H_2O_2$). $H_2O_2$ is an oxidizing agent, but it does not have an unpaired electron, so it is less reactive and much more stable than superoxide. The relative stability of $H_2O_2$ lends to its ability to serve as a cellular signal of oxidative, or redox, state, a role that is discussed later.

The hydroxyl radical is another ROS, the neutral form of the hydroxide (OH) ion. It can result from $H_2O_2$ reacting with iron, or from superoxide reacting with $H_2O_2$ [2]. Like the superoxide radical, this molecule is extremely reactive and unstable. In fact, it is more reactive than any other known radical, reacting with nearly any molecular entity typically within two molecular diameters [3].

**Biological Sources of Reactive Oxygen Species (ROS) and endogenous antioxidants**

Some production of ROS appears to be an inescapable byproduct of cellular oxygen consumption, and radicals are produced in several commonly occurring biological reactions with many being quickly neutralized by various forms of antioxidants inside the cell. For example, superoxide radicals are produced in reactions of the electron transport chain (ETC) within the mitochondria. The reduced forms of nicotinamide adenine dinucleotide (NADH) and flavin adenine dinucleotide (FADH$_2$) deliver electrons to the ETC. at specific sites, and the electrons move down the chain from carrier to carrier, releasing energy that is used to pump protons from the mitochondrial matrix to the inter-membrane space and thereby developing an electrochemical gradient that drives adenosine triphosphate (ATP) synthesis by ATP synthase. In the process of electrons being passed down the ETC., some escape, or leak, and are delivered to molecular oxygen forming superoxide. This has been shown to happen most commonly at ETC. complex I (ROS is formed within the mitochondrial matrix) and ETC. complex III (ROS formed within the mitochondrial matrix and in the inter-membrane space) [4,5]. Basal ROS production had previously been estimated to account for between 2 and 5% of oxygen consumed, but more recent evidence suggests that it is substantially less—perhaps approximately 0.15% of consumed oxygen [6,7]—and ROS production at ETC. complex I increases by approximately 187%, and at ETC. complex III by 138% during exhaustive muscular activity [7]. Note that while this is a sizeable increase in the rate of ROS production, it may be less than one might expect given that the metabolic rate of contracting muscle fibers increases approximately 100-fold or more from resting levels [8]. This suggests that mitochondria in skeletal muscle produce ROS at a much lower rate per unit of oxygen consumption during contractile activity, and one possible explanation for that is considered later.

The increase in ROS production in the mitochondria appears to be a result of increased electron flux during muscular activity, and, as noted above, much of it is reduced by superoxide dismutase (SOD) (MnSOD is found within the mitochondrial matrix, while copper- and zinc-containing SOD [Cu/ZnSOD] is primarily in the cytosol [9]) and converted to $H_2O_2$. Some of the $H_2O_2$ is further neutralized by the activity of catalase, which converts it to water and molecular oxygen. $H_2O_2$ can also be reduced to water and oxygen by other enzyme-catalyzed reactions including glutathione peroxidase and the thioredoxin systems [10-12]. However, as is discussed later, $H_2O_2$ also serves as a signaling molecule and its oxidative nature is used to trigger events in the cell.

Another muscular source of superoxide is an enzyme called nicotinamide adenine dinucleotide phosphate (NADPH) oxidase (NOX). In the context of this discussion, one important role of NADPH involves an antioxidant. Glutathione (GSH) is a molecule that is capable of dismutating ROS, itself becoming oxidized in the process. When oxidized, it will bind with another oxidized GSH molecule, resulting in its oxidized form, GSSG. In this form, it can no longer function as an antioxidant until it is reduced, and an enzyme called GSSG reductase performs that reaction using NADPH as a cofactor and producing two reduced GSH molecules that can again quench ROS
is considered to be the primary requisite substrate for eNOS results in an increase in blood flow distribution to support muscular activity. Under optimal conditions (i.e., when all necessary substrates are available), eNOS produces NO, which then induces vasodilation via relaxation of vascular smooth muscle cells resulting in hyperemia to active muscle fibers. However, the lack of any requisite substrate for eNOS results in an ‘uncoupled’ enzyme, and in this state eNOS production of superoxide is increased [26-28]. The increase in ROS has the capacity to reduce functional NO via two related mechanisms: (1) NO can react with superoxide, which results in the formation of peroxynitrite radicals; and (2) ROS, and especially peroxynitrite, can reduce the availability of one necessary substrate of NOS, thereby further uncoupling NOS and further increasing superoxide production [26,29]. These outcomes impair the capacity for hyperemia to active skeletal muscle to support contractile activity. In addition, peroxynitrite is a particularly reactive molecule, and has been shown capable of causing oxidative injury and of triggering necrosis and apoptosis [30].

A number of additional mechanisms of ROS production are known to exist, but it appears that those described here are the most productive [31], so we confine our further discussions to these pathways. Importantly, all ROS species produced in skeletal muscle are derived from either superoxide or NO [32], and the production of, and neutralizing of, these two molecules explains most of cellular redox state at any given time.

### ROS as critical signaling molecules

As discussed above, there are a number of enzyme-catalyzed pathways through which ROS are produced. While there is considerable interest in the role that oxidative stress plays in the etiology of chronic diseases, consideration of these reaction pathways is also suggestive that a threshold level of ROS must be needed for optimal cellular function. That is especially apparent when considering NOX, as the sole known purpose for this enzyme is the production of ROS, as opposed to the other reaction pathways that have been identified wherein ROS production is a by-product.

For a molecule to function effectively in a signaling role, its levels must be able to be regulated (i.e., adjusting rates of synthesis and degradation), it must have specific targets or receptors, and its signaling effect must be reversible [33]. Superoxide and hydroxyl radicals and peroxynitrite are highly reactive and unstable, and, as such, are unsuitable in a signaling role. That is, they simply cannot travel far without oxidizing some nearby molecule. However, $\text{H}_2\text{O}_2$ is considered to be the primary intracellular redox signaling molecule [34-36]. $\text{H}_2\text{O}_2$ is small and can diffuse easily, and it is stable enough to travel and oxidize specific sites on target molecules [37]. As discussed previously, superoxide and other more reactive radicals are partially neutralized by being converted to $\text{H}_2\text{O}_2$, allowing this molecule to be reflective of the overall oxidative state in the cell.

Several studies have produced results in support of ROS having a signaling role, both with short-term exercise and as a result of exercise training. As noted above, NOX-induced ROS contribute to increasing calcium release from SR. Reid [38,39] showed that reduction of...
ROS below basal levels reduced the force production of skeletal muscle, while increasing ROS to a certain point increased force production, and ROS exposure above that point resulted in a reduction in force production [40]. Other studies have demonstrated that beneficial training adaptations are eliminated when trainees are treated with antioxidants. Gomez-Cabrera et al. [41] found that 1,000 mg per day of vitamin C supplementation attenuated increased PGC-1α expression and, consequently, attenuated the increase in mitochondrial density that commonly occurs with aerobic exercise training. In addition, the training-induced increase in several antioxidant enzymes was inhibited with vitamin C supplementation [41]. Similarly, Ristow et al. [42] found that daily supplementation with a combination of 1,000 mg of vitamin C and 400 IU of vitamin E eliminated the training-induced increases in insulin sensitivity and PGC-1α expression, as well as levels of endogenous antioxidants SOD and glutathione peroxidase. It is noteworthy that another recent study of the effects of antioxidant supplementation on training adaptations in humans found that daily supplementation with 500 mg of vitamin C and 400 IU of vitamin E had no effect on physiological responses to 12 weeks of aerobic exercise training [43]. These contradictory findings may be attributable to the dosage of antioxidant supplement utilized, but further research in this area is merited [43].

Although how ROS signaling induces adaptations to training is not fully understood, a number of possible mechanisms have been identified. For example, certain proteins, especially those composed of sulfur-containing amino acids such as cysteine, are readily and reversibly oxidized by ROS [44,45]. In addition, the oxidation of amino acid residues containing aromatic structures (e.g., histidine, tyrosine, tryptophan, and phenylalanine) can result in protein conformational changes that alter affinity to target molecules and potentially increase or decrease binding to inhibitors or stimulators [46]. ROS have also been shown to affect the catalytic rate of kinases and phosphatases, a finding suggesting that ROS have the capacity to ‘fine-tune’ the magnitude and duration of inhibitory and stimulatory signals [46]. Together, these findings suggest a critical role for ROS in improving muscle function acutely, and in stimulating adaptations to exercise training, and that some critical level of ROS production may be needed to limit oxidative damage.

**ROS and oxidative stress**

Under conditions of chronic abundance of oxidants relative to antioxidants, oxidative damage can occur. Primary targets of excessive ROS are proteins, lipids, and DNA. As mentioned earlier in this review, oxidizing agents acquire electrons from target molecules. In proteins, this can result in conformational changes, some of which are irreversible. For example, the formation of carbonyl groups on several amino acid residues is an irreversible modification, and affected proteins are targeted for degradation [47,48]. Oxidative stress, then, is a stimulus for protein catabolism.

Lipid molecules can also be targeted by ROS, resulting in them being oxidized, a process called peroxidation. The interaction of a lipid molecule with a radical sets off a self-propagating chain reaction [49], as one radicalized lipid molecule interacts with others, a process that continues until two radicals interact with each other. This process is damaging to cells and tissue, and has been implicated in the development of atherosclerosis, asthma, Parkinson’s disease, and other conditions [49].

Oxidative damage to DNA can take many forms, dependent on the portion of the DNA structure affected, and the specific ROS involved [50-52]. The details of how oxidation affects DNA is beyond the scope of this review, but, briefly, oxidative damage to DNA has been linked to cancer, with its primary role likely being the initiation of carcinogenesis rather than cancer promotion/progression [52].

**Sedentariness, exercise, and oxidative stress**

The association between sedentary behavior and chronic disease is well-established, and it may be that oxidative stress is a key process whereby inactivity leads to a number of chronic diseases. Specifically, oxidative stress is known to be associated with aging, cardiovascular disease, and sarcopenia, and it may be that exercise exerts its beneficial effects on each of these disease processes through its effects on redox status.

As discussed previously, the mitochondria are one source of ROS, as superoxide is produced when electrons leak from the ETC. and bind with molecular oxygen. Mitochondrial superoxide production does seem to be increased during exercise relative to a basal state [7], and while factors such as temperature [53] may also influence mitochondrial function and ROS production, it has been suggested that long-term sedentariness produces an environment conducive to high basal levels of superoxide formation. It was noted earlier that mitochondrial ROS production increases less than threefold between rest and exhaustive exercise, while metabolic rate increases approximately 100-fold. Several studies have reported that state IV mitochondrial respiration (resting) is more productive of ROS than state III mitochondrial respiration (exercise) [53-57]. Fisher-Wellman and Neuffer [58] explain that the rate at which electrons flow down the ETC. is influenced by energy demand, and not energy availability. As such, sedentariness and...
the relative lack of adenosine diphosphate (ADP) availability results in a slower transmission of electrons through the ETC., thereby allowing for increased electron leakage. In addition, the concentration of reducing equivalents (e.g., NADH and FADH₂) influences the cumulative redox state, and an increase in the concentration of reducing equivalents as may be caused by over-nutrition induces a more reduced environment, which subsequently increases the driving force for electron entry into the ETC. at complex 1 (NADH/NAD+) and complex II (FADH₂/FAD+) [58]. As a result, though oxygen flux through skeletal muscle is many times higher during exercise, the rate of electron flow through the ETC. is faster due to the increased energy demand (state III respiration). This condition ‘de-stresses’ the mitochondria and permits less time for electron leakage, and as a result mitochondrial ROS production increases relatively little.

People who are chronically inactive also exhibit lower levels of endogenous antioxidants [59-61]. In fact, it appears that the regular inducement of exercise-mediated ROS formation stimulates an adaptation in skeletal muscle that enhances its capacity to manage oxidative stress, a finding consistent with hormesis, the concept that many biological systems respond in a bell-shaped or inverted U-shaped pattern to potential stressors [59]. For example, moderate exercise training alters oxidative homeostasis by reducing basal levels of oxidative damage [60], and by increasing resistance to oxidative stress, mediated largely through increased endogenous antioxidant defenses [60,62]. Jackson et al. [63] found that skeletal muscle contraction-induced ROS activates several identified redox-sensitive transcription factors, including nuclear factor-κB (NF-κB). This transcription factor regulates expression of cell-protective genes during adaptation to exercise [64].

Aerobic exercise training sufficient to improve cardio-respiratory fitness might also be expected to improve oxidative balance via its effects on purine metabolism. As noted previously, hypoxanthine is the substrate for XO, an enzyme that produces xanthine and superoxide. Intensive, and especially anaerobic, exercise has been shown to lead to increased XO activity [25]. Because aerobic exercise training typically increases aerobic capacity and work rate at lactate threshold, it is capable of reducing hypoxanthine levels in the plasma [65], thereby providing less substrate for XO and resulting in less ROS production at any submaximal absolute work rate.

Sirtuin 1 (SIRT1) is an enzyme that functions to remove acetyl groups from proteins involved in cellular regulation and can therefore influence expression of specific genes. Two known targets of SIRT1 are PGC-1α and FOXOs, which are Forkhead box transcription factors, some of which affect antioxidant enzyme expression (e.g., catalase and MnSOD) [66-68]. Exercise training has been shown to increase SIRT1 activity [69], and this may mediate much of the exercise-related improvement in oxidative state. Increased mitochondrial biogenesis and increased mitochondrial density alone may reduce ROS production while at rest by providing new mitochondria to supplement those that are aged or dysfunctional due to oxidative stress, as well as increasing mitochondrial membrane surface area to facilitate diffusion and dissipation of proton gradients, thereby reducing the slowing of electron flow in the ETC. during rest [68]. Additionally, long-term exercise training appears to increase the level of endogenous antioxidants in proportion to the level of ROS production, resulting in a supercompensation that affords protection from their potentially damaging effects [60,62,63].

Good evidence exists showing an inverse relationship between measures of aerobic fitness and oxidative stress [70-72], and, as has been noted, it is likely that much of the protection against oxidative stress afforded by exercise is mediated through improvements in antioxidant capacity. Mitochondrial ROS production is proportional to oxygen flux [6,7], and exercise increases ROS formation in the ETC. Likewise, ROS formation by NOX is also elevated during exercise [18], and it may be that an increase in aerobic capacity may also increase the capacity for ROS formation via these pathways. Conversely, given that improved aerobic fitness would be expected to reduce AMP levels at any given absolute exercise intensity, it may be expected that ROS synthesis by XO would be reduced with an increase in aerobic fitness. It appears that ROS production is increased during exercise, and, as such, it may be that an increase in aerobic capacity is accompanied by an increased capacity for ROS production. However, it also appears that regularly inducing an oxidative environment with exercise results in significant improvements in antioxidant levels and activity. Falone et al. [73] found that recreational runners exhibited greater serum antioxidant capacity and reduced protein carbonyl content before and after exhaustive exercise than untrained subjects. Interestingly, they also reported that, compared to pre-exercise baseline levels, exhaustive exercise resulted in reduced levels of thiobarbituric acid-reactive substances (TBARS) in trained subjects, while resulting in increased levels of TBARS in untrained subjects. (TBARS are formed as a byproduct of lipid peroxidation.) These findings support the conclusion that greater levels of aerobic fitness are accompanied by improved antioxidant capacity such that, though ROS formation is elevated with exercise, indicators of oxidative stress are reduced.

Although it appears that skeletal muscle exhibits the largest exercise-induced enhancement in controlling oxidative stress, other tissues also significantly enhance antioxidant defense mechanisms. For example, Azizbeigi et al. [74] reported that 8 weeks of progressive resistance
training improved the antioxidant capacity of erythrocytes mainly through increases in SOD activity. Similarly, Metin et al. [75] reported that erythrocytes in young male soccer players exhibited greater SOD activity than controls. While there may be numerous consequences of these adaptations in the blood (e.g., less lipid oxidation in plasma), one effect that has been identified is that erythrocytes demonstrate better resistance to ROS-induced hemolysis after exercise training [76].

Another example is the liver. This organ has a high metabolic rate at rest, but it is markedly reduced during exercise as blood flow is redirected to the working muscles [77]. As such, a relative ischemia is induced in the liver during exercise. This stimulus results in xanthine dehydrogenase being converted to XO and increased ROS production via the enzyme-catalyzed mechanism described earlier. It is noteworthy that the liver contains much higher levels of XO than does skeletal muscle [58], and this pathway therefore provides a significant proportion of hepatic ROS. Radak et al. [78] showed that in rats, exercise training at 75% of maximum oxygen uptake ($\text{VO}_{2\text{max}}$) for 8 weeks resulted in a significant reduction in ROS production in the liver compared with non-exercised animals. They also reported that reduced GSH levels were higher, and oxidized GSH (GSSG) levels were lower in the trained group than in the sedentary group, reflecting a more favorable redox state.

Redox balance in the brain also appears to respond favorably to exercise training. Regular physical exercise results in an increase in the content of several antioxidants in the brain, including Cu/ZnSOD and glutathione peroxidase, as well as increases in PGC-1a [79]. In addition, exercise appears to stimulate an increase in the production of brain-derived neurotrophic factor (BDNF) [80], though the mechanisms are not fully understood. BDNF is a uniquely versatile substance, and has been shown to influence brain development, neuroplasticity, neurogenesis, and neuron cell survival [81], and BDNF content in the brain has been shown to increase in response to exercise and oxidative stress [82]. It appears that oxidative stress stimulates an increase in BDNF content, which subsequently stimulates increased antioxidant enzyme expression, at least in part via increased NF-κB activation [83,84]. Oxidative stress has been implicated in several disorders of the brain, including Alzheimer’s disease and Parkinson’s disease [85]. A number of studies have found that regular activity provides protection against acquiring these conditions, and the anti-oxidative influence of physical exercise on the brain may provide one important explanation for those findings [85,86]. Importantly, increases in BDNF production appear to depend largely on increased neural activity, which can be induced with exercise, intellectual activity, or reduced caloric intake [87].

**Exercise-induced oxidative environment, ROS signaling, and protein synthesis**

It was noted earlier that exercise acutely increases ROS production via increased activity of several enzyme-
catalyzed processes. It was also previously noted that several studies have shown that blocking ROS formation associated with exercise also attenuates many of the key adaptations to exercise training. Taken together, these findings suggest that ROS formation is not simply a toxic byproduct of muscle contraction or oxygen metabolism, but is a key participant in stimulating beneficial adaptations to training, and, as such, regularly inducing an oxidative redox state may be important in promoting health. Several recent studies have yielded results that seem to support this position. Nyberg et al. [61] measured arteriovenous GSH and GSSG differences in the legs of young (23 ± 2 years) sedentary, old (66 ± 2 years) sedentary, and old (62 ± 2 years) active men during knee extension exercise. As noted previously, GSH is an antioxidant, and is converted to GSSG in its reduced form, so an increase in GSSG and/or an increase in the GSSG:GSH ratio would indicate a more oxidative redox state. Young sedentary men demonstrated an exercise-associated increase in GSSG and the GSSG:GSH ratio in venous blood that was correlated with intensity. Interestingly, the exercise-related increases did not occur in older sedentary or older active men, though the older active men did display higher levels of GSH and antioxidant enzymes. It appears that the oxidative stress associated with aging chronically activates NF-κB and results in increased antioxidant enzyme activity, but the capacity of aged muscle to stimulate a further increase of ROS during contraction sufficient to measurably alter the GSSG:GSH ratio is significantly reduced compared with young muscle [88-90]. Given the important signaling roles played by ROS (e.g., stimulation of mitochondrial biosynthesis, described previously), it is possible that a reduced capacity to stimulate an oxidative redox state via muscle contraction may contribute to aging and the development of chronic disease.

Recently, it has also been speculated that the oxidative environment that is associated with exercise may be critical for the synthesis of certain proteins, specifically those that exhibit disulphide bonds, many of which are membrane or secretory proteins [91]. Oxidation of cysteine residues leads to the formation of sulfenic acid intermediates, and in close proximity to other oxidized cysteine residues can result in disulphide bonds [92,93]. Said another way, these bonds cannot be formed in a reduced redox state. In fact, Ron and Harding [94] found that the endoplasmic reticulum of sedentary, insulin-resistant rats had significantly fewer disulphide bonds and, therefore, more unfolded proteins than their active, insulin-sensitive counterparts. Therefore, it may be that the transient and significant increase in ROS production and the consequent oxidative shift that results during exercise is necessary to optimize the production of these proteins, and people who are chronically inactive may not provide that environment frequently enough for optimal protein synthesis. If this is true, it provides yet one more additional mechanism whereby regular physical exercise sufficient to produce an oxidative redox state optimizes health.

Summary
Oxidative stress is a consequence of the presence of too many oxidants relative to antioxidants, and can result in cellular or tissue damage, illness, and disease. However, ROS are not entirely toxic, and research evidence indicates important healthy roles for oxidants, including cell signaling, adaptation, and inducing an environment conducive to synthesis of proteins containing disulphide bonds. Several studies report that many of the beneficial adaptations to exercise training are reduced or eliminated when ROS levels are artificially reduced, further supporting the conclusions that regularly inducing an oxidative redox state through exercise is healthful. Habitual sedentariness, especially in combination with overnutrition, may lead to chronically stressed mitochondria and increased mitochondria-sourced superoxide production, and to reduced levels of endogenous antioxidant enzymes, as well as reduced mitochondrial biosynthesis, all of which foster a vicious cycle that results in an environment that is increasingly oxidative. Conversely, regular exercise appears to produce a hormesis response, whereby the stressor (exercise) induces a transient, systemically oxidative environment that stimulates adaptations enhancing the capacity of tissues throughout the body to neutralize ROS and to more effectively regulate redox balance (see Figure 1). These adaptations provide resistance to development of a number of chronic diseases, and may explain much of the exercise-related improvement in health and longevity cited in many epidemiological studies.

Competing interests
The authors declare that they have no competing of interests.

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