Biochemical Composition and Anti-anemic Potential of *Solanum torvum* (Solanaceae) Berries in Albino Wistar Rats

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ABSTRACT---- *Solanum torvum* (Solanaceae) is a food plant commonly used in some regions of Côte d’Ivoire to treat anaemia. This study was therefore conducted to highlight some biochemical constituents and the anti-anemic potential of cooked berries of this plant in Wistar rats. To this end, the berries were cooked for 30 minutes and then the constituents were evaluated using standard biochemical methods. For the study of the anti-anemic potential, four (4) groups of eight (8) rats, aged 12 weeks and weighing on average 105 g were used. The rats were made anaemic by phenylhydrazine and then received 1ml/100g/d of distilled water (negative control), the aqueous extract of the berries at concentrations of 3.2 and 6.4 mg/ml. The positive control batch of rats received Vitafer (reference drug for the treatment of anaemia). The anaemia was assessed by means of a haemogram performed on blood samples taken on days 0, 3, 7 and 15. The analysis showed that the berries had a moisture content of 12.085%. The dry matter content was 87.915% with an ash content of 16.310%. Vitamin C was estimated at 14.810 mg/100g.

**Keywords**---- *Solanum torvum*, biochemical composition, anti-anemic, rat

1. INTRODUCTION

The use of wild plants for therapeutic purposes dates back thousands of years. In Africa, the use of traditional medicine and pharmacopoeia is a common and ancient practice. WHO [30] estimates that more than 80% of the world's population uses plants for health care. According to Faye and Champey [11], about 50% of the molecules used in human medicine that are marketed for the treatment of cancers are extracted from plants or derived from them. In Côte d'Ivoire, more than five thousand plant species have been inventoried [1], including 761 medicinal species and 1421 recipes [3]. Among these numerous plant species with therapeutic properties is *Solanum torvum* (*S. torvum*), a food plant of the Solanaceae family. *S. torvum* is a wild aubergine whose fruits and leaves are used in traditional medicine to treat several ailments [4]. Indeed, studies carried out on the organs of *S. torvum* have shown, among other things, its effective action in the treatment of malaria [6]. Moreover, the moderate inhibitory action of the fruits of this plant on alpha-glucosidase means that it could be used as an anti-diabetic [27]. The methyl caffeate isolated from *S. torvum* is reported to possess a hypoglycaemic effect and could be developed into a potent oral anti-diabetic drug [12]. In addition, *S. torvum* is reported to inhibit the growth of *Helicobacter pylori* in human gastric epithelial cells [15] and to have action against *Escherichia coli*, *Neisseria gonorrhoeae* and *Candida albicans*. According to Ramamurthy et al. [25] the fruit of *S. torvum* is an excellent source of natural antioxidants and could be an effective nutritional supplement. The methanolic extract of *S. torvum* seeds and its ethyl acetate fraction have shown antidepressant activity in humans [19]. In several regions of Côte d'Ivoire, *S. torvum* berries are used empirically to treat anaemia. However, despite the virtues attributed to the berries of this plant, few scientific studies have demonstrated their anti-anemic potential in an animal model. It is in this context that this study aimed to investigate the biochemical composition and anti-anemic potential of *S. torvum* berries in the Wistar albino rat.
2. MATERIALS AND METHODS

2.1. Collection of plant material

*S. torvum* berries were collected in a forest at Diegonefla in the department of Oumé, a town about 260 km from Abidjan in central western Côte d'Ivoire. These berries were collected three weeks after flowering from trees over one metre tall. The berries were stored in a cooler and then transported to Abidjan.

2.2. Animal material

Thirty-two (32) rats of the species *Rattus norvegicus* of the Wistar strain, including 16 males and 16 females, aged 8 weeks with an average body mass of 105 g were used for the experiment. These rats were acclimatised for one week to the rearing conditions of the Physiology, Pharmacology and Pharmacopoeia laboratory of the Nangui Abrogoua University (Côte d'Ivoire). They had free access to water and food.

2.3. Preparation of aqueous extracts of *S. torvum*

*S. torvum* berries were removed from the stems, weighed and washed. They were then cooked in water for 30 minutes at 100°C in the proportions described by Agbemafle et al. [2]. After cooking, the mixture (berries and water) was ground with an immersion blender. Part of the resulting grind was used for biochemical assays to be performed on the fresh sample and the other part dried. Ten grams (10 g) of dried sample was then dissolved in 0.3L of distilled water. The resulting mixture was homogenised in a magnetic stirrer for 24 hours. The resulting homogenate was successively filtered twice on cotton wool and once on Wattman No. 1 filter paper. The filtrate was then freeze-dried and stored for experiments.

2.4. Biochemical analysis of *S. torvum* berry extracts

The determination of dry matter and crude ash were determined by AOAC methods [5]. Iron, zinc, Magnesium and potassium were determined according to the method described by CEAEQ [7] using argon plasma ionizing source mass spectroscopy (ICP-MS). The minerals were atomized and ionized in argon plasma and the ions produced were analyzed by the spectrometer. The concentration of minerals in the sample was determined by comparison with standard solutions. The determination of vitamin C was carried out according to the method described by Pongcraz et al. [24]. The measurement of DPPH radical scavenging activity was carried out according to the method of Ranarivelo et al. [26]. The reduction of DPPH radicals was determined by measuring the absorption at 517 nm. The radical scavenging activity was calculated as a percentage of DPPH discoloration using the following equation:

\[
\text{DPPH radical scavenging } \% = \left( \frac{A_0 - A_1}{A_0} \right) \times 100
\]

Where \( A_0 \) is the absorbance of the DPPH solution and \( A_1 \) is the absorbance of the sample.

2.5. Experimental animals

2.5.1. Induction of Anaemia

Anemia was induced by phenylhydrazine Chloridrate. Phenylhydrazine was previously dissolved in a DMSO solution diluted to 10/100 in distilled water. It was administered to rats intraperitoneally at a dose of 40 mg / kg of body weight / day [21] for two days.

2.5.2. Treatment of the Animals

Two days after induction of anaemia, rats were treated orally according to the following protocol [22]:

- Group 1: (negative control rats) rendered anaemic received 1 ml/100g body weight/day of distilled water ;
- Group 2: (positive control rats) rendered anaemic, received 1 ml/100g body weight/day of Vitafer, a pharmaceutical reference anti-anaemic;
- Group 3 (test batch) anaemic rats were given by gavage 1 ml/100g body weight/day of the aqueous extract of *S. torvum* berries at a concentration of 3.2 mg/ml;
- Group 4 (test batch) of anaemic rats received by gavage 1 ml/100g body weight/day of aqueous extract of *S. torvum* berries at a concentration of 6.4 mg/ml.

To evaluate the effect of *S. torvum* extracts on anaemia, haematological parameters such as red blood cells, haemoglobin and haematocrit were studied on each blood sample collected on days 0, 2, 7 and 15 by an automatic analyser URIT- 3000 Plus (Guangzhou, China).
2.6. Statistical analysis

Data were collected from the analysis of biochemical parameters of cooked berries of S. torvum with 3 replicates. The results were expressed as mean ± standard deviation. XLSTAT 7.1 software was used to process the data using analysis of variance (ANOVA) followed by a Tukey post hoc test. Differences were considered significant if p < 0.05. GraphPad prism 5.0 software was used to produce the graphs for free radical scavenging activity. Dunnett’s test (α=0.05) was performed to compare the IC$_{50}$ of S. torvum with that of the vitamin C control.

3. RESULTS AND DISCUSSION

3.1. Biochemical analysis of S. torvum berry extracts

The results of the chemical composition of cooked and dried S. torvum berries are presented in Table 1. The analysis shows that the moisture content of the berries is 12.085%. The dry matter content was 87.915% with an ash content of 16.31%. The mineral values are 4.175 mg/100g for iron, 1.915 mg/100g for zinc, 1.245 mg/100g for magnesium and 134.78 mg/100g for potassium. The high ash content indicates a high mineral presence. Indeed, ash represents the mineral fraction of food. The iron content is about the same as that reported by Grubben et al. [13] in S. torvum berries, i.e. 4.175 mg/100g. However, Kouadio et al. [16] obtained a higher iron content in berries cooked for 25 minutes compared to our work. This variation in iron content could probably be explained by the growing conditions but also the harvesting period [23]. On the other hand, some authors have shown that cooking increases the majority of minerals [14]. The potassium content is higher than in Solanum aethiopicum fruits (47 mg/100g). This could be due to the growing conditions. Moreover, potassium is the most abundant mineral in plant products. The berries of S. torvum contain zinc and magnesium, two minerals that are essential for the proper functioning of the body. Indeed, zinc is involved in growth and magnesium in the synthesis of enzymes necessary for energy release and the synthesis of protein enzymes [17]. The content of zinc and magnesium are however lower than those revealed by Dan et al. [9] in Solanum aethiopicum berries which is 5.26 mg/100g for zinc and 404.64 mg/100g for magnesium.

Vitamin C indicated a level of 14.81 mg/100g fresh matter. The results of the evolution of the free radical scavenging activity as a function of concentration (Figure 1) showed that the percentage of inhibition increased as the concentrations of vitamin C and S. torvum increased. The 50% inhibitory concentrations of the DPPH radical (IC$_{50}$) were determined graphically. The IC$_{50}$ of vitamin C (1.67 µg/ml) does not differ significantly from that of S. torvum (1.87 µg/ml). Thus, the free radical scavenging capacity of S. torvum (53.47 µmol/µg.ml) does not differ significantly from that of vitamin C (60.24 µmol/µg.ml) (Table 2). The level of vitamin C in S. torvum berries is less than 15 mg /100g. This could be due to the fact that the berries were cooked. Vitamin C is a water-soluble and heat-labile vitamin. It could therefore have been destroyed by the heat. However, this level is higher than that reported by Grubben et al. [13] in raw S. torvum berries, i.e. 4 mg /100g. Antioxidant activity is high in S. torvum berries. This is shown by an IC$_{50}$ like that of the vitamin C control. S. torvum berries therefore have good antioxidant potential with a high antiradical power. Indeed, this antiradical power is higher than that of steamed potato leaves which is 12.85 µmol.ml/µg.

| Parameters       | Quantity        |
|------------------|-----------------|
| Dry matter (%)   | 87.915 ± 0.06   |
| Moisture (%)     | 12.085 ± 0.06   |
| Ash (%)          | 16.31 ± 3.86    |
| Iron (mg/100 g)  | 4.175 ± 0.035   |
| Zinc (mg/100 g)  | 1.915 ± 0.007   |
| Magnesium (mg/100 g) | 1.245 ± 0.007 |
| Potassium (mg/100 g) | 134.78 ± 0.36  |
| Vitamin C (mg/100 g) | 14.81 ± 2.563  |

Each result is expressed as mean ± standard deviation (n=3)
Table 2. Inhibitory, efficient and free radical scavenging concentration of vitamin C in relation to S. torvum berries

| Parameters     | IC$_{50}$ (µg/ml) | EC$_{50}$ (µmol/mg) | PAR (µmol.ml/µg) |
|----------------|---------------------|----------------------|------------------|
| Vitamin C      | 1.67 ± 0.76         | 0.0167               | 59.88            |
| S. torvum      | 1.87 ± 0.53         | 0.0187               | 53.47            |

IC$_{50}$: Inhibitory concentration  
EC$_{50}$: Effective Concentration  
PAR: Anti-radical power

3.2. Animal experiments

3.2.1. Effect of aqueous extract of cooked S. torvum berries on red blood cell count in rats

The results (Table 3) indicated an effective decrease in red blood cells after induction of anaemia in all experimental groups on day 3 of the study. With the exception of the negative controls, the other groups showed a very significant erythrocyte recovery on days 7 and 15 of the experiments. The aqueous extract of cooked S. torvum berries at concentrations of 3.2 and 6.4 mg/mL, respectively, resulted in similar recovery rates to the reference product on day 7 of the experiments (+65.49% and +77.70% vs +76.11%). In addition, the aqueous extract of cooked berries resulted in greater erythrocyte recovery on day 15 at concentrations of 3.2 and 6.4 mg/mL compared to Vitafer (+116.43% and +223.64% vs +100%) (Table 3). These results are broadly similar to those obtained by Loukou et al.[18] who showed that administration of 1000 mg/kg/d of Justicia Galeopsis leaf extract cooked for 30 min increased erythrocyte with a recovery of +98.11%. The extract of leaves cooked for 30 min allows a better stimulation of red blood cell synthesis. According to Elliott et al.[10], the presence of iron is essential for erythropoiesis, contributing to the production of red blood cells in the later stages of erythroid differentiation.

Table 3. Effect of aqueous extract of S. torvum berries on erythrocyte in rats

| Group (Treatment) | Variation | Erythrocyte (10$^9$/µL) |
|-------------------|-----------|-------------------------|
|                   | D0        | D3          | D7          | D15         |
| Negative controls | 4.96±2.50 | 2.67±0.24  | 2.23±1.36  | 3.46±1.38  |
| Variation         | -38.42% a | -16.48% b  | +29.58% b** |            |
| 1 ml/kg/d of vitafer | 5.94±2.43 | 2.47±1.20  | 4.53±0.38  | 4.94±0.43  |
| Variation         | -58% b**  | +76.11% b** | +100% b**  |            |
| 3.2 mg/mL         | 5.81±0.98 | 2.13±1.40  | 4.59±1.04  | 4.51±0.23  |
| Variation         | -63.33% b** | +65.49% b** | +116.43% b** |            |
| 6.4 mg/mL         | 5.42±1.07 | 1.48±1.10  | 2.63±1.14  | 4.79±0.16  |
| Variation         | -72.69% b** | +77.70% b** | +223.64% b** |            |

a: percentages of change from day D0; b: percentages of change from day D3; *: Significant difference (p <0.01); **: Highly significant difference (p <0.01)

3.2.2. Effect of aqueous extract of cooked S. torvum berries on haemoglobin levels

The results (Table 4) showed a highly significant decrease in haemoglobin levels in all groups after induction of anaemia with phenylhydrazine. After treatments, an increase in haemoglobin was observed from day 7 and progressed to day 15 compared to the negative control group. However, the aqueous extract of cooked S. torvum berries at the concentration of 6.4 mg/mL resulted in a higher recovery rate on day 15 compared to the reference anaemia drug (95.19% vs 77.12%) (Table 4). Our results are similar to those obtained by Tchogou et al.[28] and Loukou et al.[18] who observed in their study a higher haemoglobin recovery rate in the batches of rats treated with aqueous extract of Coco nucifera (Areaceae) and Justicia Galeopsis (Acanthaceae) respectively compared to the control treated with Vitafer. It is known that iron is part of haemoglobin, myoglobin and cytochrome. The presence of a high content of vitamin C could contribute to a significant absorption of non-haem iron from the aqueous extract of cooked S. torvum berries; reducing the ferric ion to a ferrous form or forming a soluble complex in the small intestine, which would increase or enhance haemoglobin formation [8].
Furthermore, the presence of zinc in the aqueous extract of cooked *S. torvum* berries may play a major role in haemoglobin synthesis. Zinc deficiency has been shown to be associated with anaemia and erythrocyte fragility [20]. All the micronutrients revealed in our study in the aqueous extract of cooked *S. torvum* berries would justify the correction of phenylhydrazine-induced anaemia.

Table 4. Effect of aqueous extract of cooked *S. torvum* berries on haemoglobin levels

|                          | Hemoglobin (g/dL) |       |       |       |
|--------------------------|-------------------|-------|-------|-------|
|                          | D0                | D3    | D7    | D15   |
| Group 1 (Negative controls) | 11.30±0.98        | 6.40±1.51 | 7.75±3.04 | 10.01±0.87 |
| Variation                |       | -43.36%*  | +21.09%b | +57.81%b* |
| Group 2 (1ml/kg/d of vitafer) | 13.60±1.02        | 7.17±2.57 | 11.4±0.71 | 12.7±0.60 |
| Variation                |       | -47.27%*  | +58.99%b* | +77.12%b* |
| Group 3 (3.2 mg/mL)      | 13.60±1.48        | 7.65±3.39 | 11.52±1.83 | 13.52±1.38 |
| Variation                |       | -43.75%*  | +50.58%b* | +76.73%b* |
| Group 4 (6.4 mg/mL)      | 10.80±1.17        | 5.62±2.45 | 9.52±2.19 | 10.97±0.99 |
| Variation                |       | -38.70%*  | +69.39%b* | +95.19%b* |

a: Percentage change from day D0; b: Percentage change from day D3; *: Significant difference (p <0.01); ** : Highly significant difference (p <0.01)

3.2.3. Effect of aqueous extract of cooked *S. torvum* berries on haematocrit levels in rats

Table 5 shows the effect of the treatments on haematocrit levels in rats. The aqueous extract of cooked *S. torvum* berries, as well as Vitafer, significantly increased haematocrit recovery in treated rats (Table 5). The observed decrease in haematocrit was fully corrected in rats given the cooked berry extract with a recovery rate of over 100%. This recovery rate is higher than that obtained by the batch of rats given Vitafer and those given distilled water. These observations are consistent with the results of the red blood cell and haemoglobin levels in this study.

Table 5. Effect of aqueous extract of cooked *S. torvum* berries on haematocrit level

|                          | Hématocrite (%) |       |       |       |
|--------------------------|-----------------|-------|-------|-------|
|                          | D0              | D3    | D7    | D15   |
| Group 1 (Negative controls) | 33.9±2.95        | 22.2±4.53 | 23.25±6.82 | 29.57±2.29 |
| Variation                |   -34.51%*     | -4.72%b | +33.19%b* |
| Group 2 (1ml/kg/d of vitafer) | 40.8±3.07        | 21.52±7.73 | 34.20±2.15 | 36.32±2.71 |
| Variation                |   -47.25%*     | +58.92%b* | +68.77%b** |
| Group 3 (3.2 mg/mL)      | 40.8±4.47        | 22.95±7.18 | 34.57±5.52 | 40.35±4.42 |
| Variation                |   -43.87%*     | +50.63%b* | +75.81%b** |
| Group 4 (6.4 mg/mL)      | 32.4±3.52        | 19.87±7.37 | 31.57±6.58 | 32.92±2.99 |
| Variation                |   -38.67%*     | +58.88%b* | +65.67%b** |

a: Percentage change from day D0; b: Percentage change from day D3; *: Significant difference (p <0.01); ** : Highly significant difference (p <0.01)

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