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Morphological, Molecular and Metabolic Characterization of the Pigmented Fungus *Subramaniula asteroids*

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Abstract: Chaetomiaceae fungi are ascosporulating fungi whose importance as human pathogens has been frequently ignored. In the current study, a new isolate of the genus *Subramaniula* was described. The fungus was isolated from the soil of Wadi Om Nefa’a, Hurghada in the Red Sea Governorate, Egypt. Previously, *Subramaniula* were misidentified as *Papulaspora* spp. According to molecular analysis, the fungus was identified as *Subramaniula asteroids* OP484336. Remarkably, this species has been found among other fungi responsible for keratitis in humans and has been recorded for the first time in Egypt. Analysing the *Subramaniula asteroids*’ metabolic profile was one of the objectives of the current study because little is known about this family’s metabolome. The fungal extract’s untargeted metabolic profiling was carried out by gas chromatography-mass spectroscopy (GC/MS), 1H and 1H-HSQC nuclear magnetic resonance (NMR) data, and their corresponding databases. In total, fifty-nine metabolites have been reported in the polar and non-polar extracts. The majority of polar metabolites are amino acids and carbohydrates. The non-polar extract’s main components were 1-dodecanamine, N,N-dimethyl-, 1-tetradecanamine, N,N-dimethyl-, and 9-octadecenoic acid ethyl ester. The current study is the first to provide a metabolic profile of *Subramaniula asteroids*, which can be used in chemotaxonomical classification, antifungal drug development, and biological activity investigation of the studied species.

Keywords: *Subramaniula asteroids*; chaetomiaceae; keratitis; metabolic profile; NMR analysis; GC/MS

1. Introduction

Melanized fungi are significant human pathogens. Human infections have been related to approximately 70 genera of this fungal group [1]. Recently, the importance of Sordariales has been emphasized, especially Chaetomiaceae, whose distribution has been neglected due to identification issues [1]. A significant number of Chaetomiaceae-like fungi fail to form traditional diagnostic structures in culture, making accurate morphological descriptions difficult. In the clinical laboratory, fungi that did not develop conidia were recently labelled as unidentifiable ‘mycelia sterilia’ [2]. *Papulaspora* was used to describe isolates that formed clumps or bulbils, whereas *Madurella* was used to describe filamentous subcutaneous infection strains [1]. Such isolates were observed to be diverse, relating to several genera and families of ascomycetes using molecular phylogenetic DNA sequencing studies. Furthermore, species were discovered that contained numerous other related sub-species that had been previously thought to belong to the same taxon [3]. Despite recent advances in the taxonomy of sterile fungi, *Papulaspora*’s phylogenetic position remains unknown. The occurrence of clumps or papulospores is non-specific and polyphyletic, as a broad diversity of fungi from different genera and orders can produce similar compact
masses of cells. Recently, there has been a rise in interest in nonsporulating chaetomium-like isolates that are an infection source in humans similar to keratitis or subcutaneous infection after trauma [4]. Despite using recent molecular techniques, researchers were unable to categorize these isolates in the genus Chaetomium at the species level due to the genus’ current state of morphological confusion and significant phylogenetic variation. Over 300 Chaetomium species have been described, and although most lack current descriptions, only a few have been defined with DNA data. For the taxonomic analysis of fungal species, the chemotaxonomic method using metabolic profiling is a high-throughput technology that supplements current molecular methods [5]. NMR structure analysis is a useful tool for studying metabolic profiling of living organisms. Because of its distinctive characteristics, including universal, repeatable, and non-destructive measurements, and the quick development of NMR software and hardware, metabolomic researchers are now able to use this tool for the immediate detection and quantification of compounds present inside any living system [6]. NMR analysis is increasingly being used in pathogenic fungus metabolite profiling. It provides accurate information on metabolites found inside the tissue as well as those produced externally, which can help differentiate between fungi of the same genus [7]. GC/MS is ideal for the investigation of the non-polar metabolites. Because of its high level of sensitivity, GC-MS may be used to analyze compounds that are present in minute concentrations [8].

The primary goal of the current study is to describe a novel fungal strain isolated from Hurghada, Wadi Om Nefa’a, Red Sea Governorate, Egypt, using molecular characterization, which was previously erroneously morphologically identified as Papulaspora species [8]. The metabolic profile of the new isolate was investigated using NMR spectroscopy and GC/MS.

2. Materials and Methods

2.1. Collection of the Studied Soil Samples

The enriched soil around cultivated plants such as Zilla spinosa, Artemisia Judaica, Cleome droserifolia, Lotus sp., and Acacia raddiana was collected from Wadi Om Nefa’a, Hurghada, Red Sea Governorate (27°4′13.07″ N, 33°24′29.30″ E), Egypt (Figure 1), in the summer of 2018. The soil was placed in the Mycology lab, Faculty of Science, Helwan University, Egypt.

Figure 1. Map showing the location of sample collection.

2.2. Fungal Isolation and Morphological Characterization

The studied strain was isolated using the serial dilution method [9]. Serial dilutions were obtained by mixing a 1 g soil sample with sterilized distilled water (9 mL) in a sterilized test tube, shaking well, and then transferring 1 mL of the soil suspension to
achieve a serial dilution of $10^{-3}$. Then, each dilution was transferred to a Petri dish with potato dextrose agar medium (PDA), then rotated to mix the soil dilution with the media before solidification of the media. The cultured Petri dishes were then stored at 25 °C for 7 days. The isolated fungal species was purified and morphologically identified as *Papulaspora irregularis* at Mycological Center, Assiut-University, Egypt (AUMC).

### 2.3. Submerged Cultivation of the Fungal Isolate

A 50 mL potato dextrose yeast extract (PDY) broth medium containing 1 L potato infusion, 20 g dextrose, and 2 g yeast extract [10] was inoculated with two fungal plugs (5 mm) of one-week-old PDA fungal culture. The fungal cultivation was performed using static incubation conditions for 7 days at 25 °C. The fungal biomass and culture filtrate were separated by filtration using Whatman filter paper. The mycelia were thoroughly washed with distilled water before being lyophilized and stored for further analysis.

### 2.4. Molecular Identification

#### 2.4.1. Extraction of Fungal DNA and Polymerase Chain Reaction (PCR) Amplification

The total genomic DNA was extracted from the fungal mycelium of one-week-old PDA culture using Quick-DNA™ Fungal/Bacterial Miniprep Kit (ZYMO RESEARCH). COSMO PCR RED Master Mix (W1020300X) (Birmingham, England) was used for the PCR reactions according to manufacturer instructions. All reactions were conducted in a reaction volume of 50 µL, with the internal transcribed spacer 1 (ITS1) and ITS4 primers used to amplify ribosomal internal transcribed spacer. The purified product using Zymo-Spin™ Technology was sequenced with Sanger technology using an ABI 3730xl DNA sequencer and ITS4 primer from Eurofins Genomics (formerly GATC Biotech; Ebersberg, Germany).

#### 2.4.2. Sequencing and Analysis

The blast was carried out for the achieved nucleotide sequence using Geneious pro program 11.1.5. Comparatively, the sequence obtained from the National Center for Biotechnology Information (NCBI) (https://blast.ncbi.nlm.nih.gov/Blast.cgi, accessed on 20 January 2019), with high similarity, was aligned using MUSCLE tools. The phylogenetic tree was constructed using maximum likelihood methods [11].

#### 2.4.3. Polar and Non-Polar Fungal Metabolites Extraction

Liquid nitrogen was used to ensure a uniform consistency of the dry biomass. Metabolites were extracted using a 2:2:1.8 ratio of methanol, chloroform, and water based on the dry weight of the sample [12]. The amount of solvent that should be applied to each sample was calculated using the dry mass and the percentage of water loss [13,14]. Cold methanol and water were added to each 20 mg sample, and the mixture was vortexed before being added to cold chloroform and water in glass tubes. The solution was placed in an ice bath for 10 min before being centrifuged at 4 °C for 10 min at 2000×g. The top layer (polar) was separated and dried in a Centrivap vacuum concentrator (Labconco, Kansas City, MO, USA) for approximately 48 h until the polar solvents were entirely evaporated, while the non-polar layer was transferred to fresh glass vials and dried using nitrogen gas.

#### 2.4.4. NMR Data Collection and Metabolite Identification

To acquire NMR data and process the mycelium polar extract, the following procedures were followed. Each sample was reconstituted in 620 µL NMR buffer composed of 100 mM sodium phosphate buffer (pH 7.3), 1 mM TMSP (internal standard, 3-(trimethylsilyl) propionic-2,2,3,3-d4 acid), and 0.1% sodium azide, in D2O. The 1D and 2D data were collected using a 700 MHz Bruker Avance™ III spectrometer. The data were gathered in 2.9 s using 64K points and a spectral width of 16.0 ppm. Overall, 120 scans, 4 dummy scans, 3 s of relaxation delay, and on-resonance pre-saturation at the residual water frequency were utilized to get the first increment of the presat-noesy spectra for solvent suppression. Topspin 2.1.1’s automatic pulse calculation experiment (pulsecal) was used to calculate the
pulse widths at 90 °C for each sample (BrukerBioSpin, Billerica, MA, USA). The Bruker hsqcedetgpsisp 2.2-pulse sequence was used to collect two-dimensional HSQC data. The $^1$H was observed in the F2 channel with a spectrum width of 11 ppm, whereas the $^{13}$C was detected in the F1 channel with a spectral width of 180 ppm.

The fungal metabolites of the polar extract were identified by comparing our $^1$H data to the Chenomx NMR Suite’s chemical library (Chenomx Inc., Edmonton, Alberta, Canada). To confirm the identities of the metabolites obtained by Chenomx, $^1$H−$^{13}$C HSQC results were compared to reference data from the Human Metabolome Database (HMDB) (https://hmdb.ca/, accessed on 23 October 2022) and the scientific literature.

2.4.5. GC/MS Data and Metabolite Identification

The non-polar extract of the fungal mycelium was analyzed with a TRACE GC Ultra Gas Chromatograph instrument (Thermo Fisher Scientific, Waltham, MA, USA) equipped with both a thermal mass spectrometer detector (ISQ Single Quadrupole Mass Spectrometer) and a TR-5 MS column (30 m × 0.32 mm i.d., 0.25 µm film thickness). The experiment was conducted with helium serving as the carrier gas at a flow rate of 1.0 mL/min and a split ratio of 1:10. After setting it at 60 °C for one min, the temperature was ramped at a rate of 4 °C/min to 240 °C and then maintained for one minute. Both the injector and the detector were kept at a temperature of 210 °C. The sample was diluted 1:10 with hexane ($v/v$) before being injected as 1.0 µL aliquot. The mass spectral range was set to $m/z$ 40–450; electron ionization (EI) was performed at 70 eV. The identification of the non-polar metabolites was performed using the AMDIS program (www.amdis.net, accessed on 23 October 2022), retention indices (relative to n-alkanes C8-C22), and mass spectrum, which were matched to authentic standards (once it becomes accessible) as well as NIST databases and Wiley spectral library.

3. Results

The fungal morphological examination was performed on potato dextrose agar (PDA). Within three days of incubation, the colonies were white and cottony, then gradually dark pigmentation was observed at the reverse surface (Figure 2A). Microscopic examination showed the cellular clump formation, which is typical of Papulaspora species (Figure 2B).

![Figure 2. The morphological (A) and microscopic (B) characterization of Subramaniula asteroids.](image_url)

Subsequently, the fungal DNA nucleotide sequence was examined using advanced Basic Local Alignment Search Tool (BLAST) (Megablast) searches from the National Center for Biotechnology Information in the GenBank database (NCBI). The fungal isolate was identified and submitted to the GenBank as Subramaniula asteroids Eg2021 with accession number OP484336. In addition, a phylogenetic tree constructed from the partial sequence and
related sequences from other fungi in GenBank demonstrated that *Subramaniula asteroids* Eg2021 is 98.2% similar to *Subramaniula asteroids* 4DR (Figure 3).

**Figure 3.** The molecular characterization of *Subramaniula asteroids*. The phylogenetic tree was drawn by Geneious pro program using maximum likelihood methods.

**NMR Spectroscopy and GC/MS-Based Metabolic Profiling**

*Subramaniula asteroids* polar extract was explored in this study by analyzing 1D NMR spectra using the Chenomx database, then comparing 2D NMR data with those found in the HMDB and literature.

Of the polar compounds, 32% are amino acids, 19% are sugars, 5.7% are alkaloids, 3% are phenols, and 2.7% are vitamins (Figure 4). The cross peaks observed in the HSQC spectrum facilitated the assignment of \(^1\)H-\(^{13}\)C correlation. The amino acids alanine and valine were identified by doublet signals at δH 1.48 ppm (3Hβ, J = 7.25 Hz) correlated with δC 18.6 and δH 1.03 (3Hγ, J = 7.00 Hz) and correlated with δC 19.5, respectively; arginine was identified with signals at δH 1.64 (2Hγ, m), δH 1.92 (2Hβ, m), and δH 3.75 (1Hα, t, J = 6.2 Hz) (Figure 5A). Sugar abundance regions were characterized by the presence of glucose, fructose, sucrose, and trehalose signals (Figure 5B). The monosaccharides glucose and fructose showed signals attributed to their anomeric protons. Additionally, the characteristic resonance positions of these protons and their corresponding coupling constants distinguished the two configurations of glucose as α- (δH 5.24, d, J = 3.68 Hz)/δC 98.9 and β- (δH 4.66, d, J = 7.88 Hz)/δC 94.8. Sucrose and trehalose were identified by the proton...
signal at $\delta H$ 5.42 (1H, d, $J = 3.89$ Hz)/$\delta C$ 95.1; and $\delta H$ 5.20 (1H, d, $J = 3.73$ Hz)/$\delta C$ 95.8, respectively. The 1H NMR spectrum revealed the presence of three signals with multiplet splitting in the aromatic region (Figure 5C) at $\delta H$ 7.31, 7.36, and 7.41 that were assigned for the aromatic amino acid phenylalanine. Moreover, tyrosine identified by the multiplet split signals resonated at $\delta H$ 6.88 and 7.18; 3.10 (dd, $J = 7.30, 14.30$ Hz), respectively. Alkaloids oxypurinol and xanthine have been annotated based on the two singlet signals at $\delta H$ 8.2 and 7.9, respectively.

Figure 4. The percentage of chemical classes identified in the polar extract of Subramaniula asteroids.

As shown in Table 1, the polar metabolites have been described with their chemical formula, molecular weight, coupling constants, and 1D and 2D NMR chemical shifts. GC/MS was utilized to characterize the chemical profile of the non-polar extract. Twenty-one metabolites were successfully identified: the main constituents were 1-dodecanamine, N,N-dimethyl-, 1-tetradecanamine, N,N-dimethyl-, and 9-octadecenoic acid ethyl ester, with peak areas of 27.96, 10.10, and 9.27%, respectively. Metabolites cyclopropanebutanoic acid, 2-[[2-[(2-pentylcyclopropyl) methyl] cyclopropyl] methyl] cyclopropyl]; phen-1,4-diol, 2,3-dimethyl-5-trifluoromethyl-methyl], methyl ester and 7,9-di-tert-butyl-1-oxaspiro (4, 5) deca-6,9-diene-2,8-dione were represented in relatively lower concentrations, with peak areas of 0.91, 0.92, and 0.93%, respectively. The non-polar metabolites are recorded in Table 2 along with their chemical formula, molecular weight, retention time, and peak area (%).
Figure 5. 1D $^1$H NMR spectra of *Subramaniula asteroids* polar extract. (A) Aliphatic region, (B) sugar abundance signals region, and (C) aromatic region.

Table 1. The identified metabolites in the polar extract of *Subramaniula asteroids*.

| Compound Name       | Chemical Formula | Weight (Da) | Assigned $^{13}$C Chemical Shift from $^1$H-$^{13}$C HSQC (2D NMR) | Assigned $^1$H Chemical Shift from $^1$H (1D NMR) | Coupling Constant |
|---------------------|------------------|-------------|-------------------------------------------------------------------|-----------------------------------------------|------------------|
| 1 3-Hydroxyisovalerate | $C_5H_9O_3$      | 118.13      | 30.6 (CH$_2$), 52.0 (CH$_3$)                                      | 1.26 (CH$_3$, s), 2.35 (CH$_2$, s)            | -                |
| 2 4-Aminobutyrate    | $C_4H_9NO_2$     | 103.11      | 26.4 (CH$_2$), 37.0 (CH$_2$), 42.0 (CH$_3$)                      | 1.90 (CH$_2$, m), 2.28 (CH$_2$, t), 2.99 (CH$_2$, t) | 7.35, 7.40       |
| Compound Name | Chemical Formula | Weight (Da) | Assigned $^{13}$C Chemical Shift from $^1$H,$^{13}$C HSQC (2D NMR) | Assigned $^1$H Chemical Shift from $^1$H (1D NMR) | Coupling Constant |
|---------------|------------------|------------|-------------------------------------------------|---------------------------------|-----------------|
| Acetate       | C₂H₄O₂           | 60.05      | 26.2 (CH₃)                                      | 1.92 (CH₃, s)                   | -               |
| Alanine       | C₃H₇NO₂          | 89.09      | 53.5 (C^α) 18.9 (C^β)                          | 3.78 (H^α, q) 1.41 (H^β, d)    | 7.2, 7.26       |
| Arginine      | C₆H₁₄N₂O₂        | 174.20     | 26.6 (C^α) 30.3 (C^β) 57.1 (C^γ)               | 1.64 (H^α, m) 1.92 (H^β, m)    | -               |
| Betaine       | C₅H₁₁NO₂         | 117.14     | 69.2 (CH₃) 56.2 (CH₂)                          | 3.90 (H^α, s) 3.26 (H^β, s)    | -               |
| Choline       | C₃H₄NO           | 104.17     | 58.6 (C^α) 56.8 (C^β) 70.4 (C^γ)               | 4.06 (H^α, m) 3.21 (H^β, s)    | -               |
| Ethanol       | C₃H₆O            | 46.06      | 119.5 (CH₃) 60.3 (CH₂)                          | 1.17 (CH₃, t) 3.63 (CH₂, q)    | 6.7, 6.7        |
| Ethanolamine  | C₃H₇NO           | 61.08      | 60.5 (CH₂)                                      | 3.8 (CH₂, t) 3.13 (CH₂, t)     | 5.21            |
| Formate       | CH₂O₂            | 46.02      | 78.2 (tCH)                                      | 4.12 (tCH, dd)                 | -               |
| Fructose      | C₆H₁₂O₅          | 180.15     | 66.1 (tCH₂)                                    | 4.02 (tCH₂, dd)                | 12.72, 1.04     |
| Fumarate      | C₄H₆O₄           | 116.07     | 138.0 (CH)                                     | 6.49 (CH, s)                   | -               |
| Gallate       | C₇H₁₀O₃          | 170.11     | 111.8 (CH)                                     | 7.00 (CH, s)                   | -               |
| Glucose       | C₆H₁₂O₅          | 180.15     | 98.9 (t^3CH) 94.8 (t^2CH) 74.4 (t^2CH)         | 4.66 (t^3CH, d) 5.24 (t^2CH, d) | 7.88, 3.68      |
| Glucose-6-P   | C₆H₁₂O₅·P        | 260.13     | 94.9 (t^3CH) 98.9 (t^3CH) 65.4 (t^2CH)         | 5.2 (t^3CH, d) 4.6 (t^3CH, d)  | 3.75, 8.0       |
| Glutamate     | C₅H₇NO₄          | 147.12     | 29.7 (C^β)                                      | 2.04 (H^β, m)                  | -               |
| Glycine       | C₂H₅NO₂          | 75.06      | 44.3 (C^α)                                      | 3.36 (H^α, s)                  | -               |
| Isoleucine    | C₄H₁₂NO₂         | 131.17     | 26.8 (C^γ) 17.5 (C^γ)                          | 1.29 (H^γ, m) 1.00 (H^γ, d)    | -               |
| Lactate       | C₃H₆O₃           | 90.07      | 22.1 (CH₃)                                     | 1.33 (CH₃, s)                  | 6.88            |
| Leucine       | C₄H₁₁NO₂         | 131.17     | 24.7 (C^δ)                                     | 0.98 (H^δ, t)                  | 6.1             |
| Methylsuccinate | C₄H₆O₄         | 132.11     | 19.7 (CH₂) 44.9 (CH₂)                          | 1.01 (CH₂, d) 2.11 (CH₂, m)    | 6.7             |
| O-Phosphocholine | C₅H₁₃NO₄·P    | 184.15     | 56.5 (CH₃) 86.9 (CH₂) 60.6 (CH₂)               | 3.20 (CH₃, s) 3.58 (CH₂, m)    | -               |
| Osypurinol    | C₅H₁₂N₂O₂        | 152.11     | -                                               | 8.20 (CH, s)                   | -               |
| Phenylalanine | C₄H₁₁NO₂         | 165.18     | 39.2 (C^β) 58.5 (C^α) 132.1 (C^β) 130.4 (C^δ) 131.8 (C^γ) | 3.10 (H^β, dd) 3.39 (H^α, q) 7.31 (H^β, m) 7.36 (H^γ, m) 7.41 (H^δ, m) | 15.4, 6.3       |
| Proline       | C₃H₅NO₂          | 115.13     | 64.2 (C^α) 49.0 (C^β) 31.9 (C^γ) 26.7 (C^δ)   | 4.13 (H^α, m) 3.41 (H^γ, m) 2.36 (H^δ, m) | -               |
Table 1. Cont.

| Compound Name | Chemical Formula | Weight (Da) | Assigned $^{13}$C Chemical Shift from 1H-HSQC (2D NMR) | Assigned $^1$H Chemical Shift from $^1$H (1D NMR) | Coupling Constant |
|---------------|------------------|-------------|-----------------------------------------------------|-------------------------------------------------|------------------|
| 27 Pyroglutamate | C$_5$H$_7$NO$_3$ | 129.11 | 27.9 ($^1$CH) 60.9 ($^3$CH) | 2.40 (H$_3^+$, m) 4.16 (H$_3^+$, dd) | 9.0, 5.9 |
| 29 Sarcosine | C$_4$H$_7$NO$_2$ | 89.09 | 35.5 (CH$_2$) 53.4 (CH$_2$) | 2.73 (CH$_2$, s) 3.61 (CH$_2$, s) | - |
| 31 Succinate | C$_4$H$_6$O$_4$ | 118.08 | 37.1 (CH$_2$) | 2.41 (CH$_3$, s) | - |
| 32 Sucrose | C$_{12}$H$_{22}$O$_{11}$ | 342.29 | 95.1 ($^1$CH) 79.8 ($^3$CH) 76.8 ($^5$CH) 73.9 ($^7$CH) 72.1 ($^9$CH) 65.3 ($^11$CH) 64.2 ($^13$CH) | 5.42 ($^1$CH, d) 4.22 ($^3$CH, d) 3.61 ($^5$CH, s) 3.48 ($^7$CH, s) 3.48 ($^9$CH, m) | - |
| 34 Trehalose | C$_{12}$H$_{22}$O$_{11}$ | 342.30 | 95.8 ($^1$CH) 73.6 ($^3$CH) | 5.20 ($^1$CH, d) 3.66 ($^3$CH, dd) | - |
| 35 Tyrosine | C$_4$H$_{11}$NO$_3$ | 181.18 | 38.2 ($^8$C$_3$) 118.9 ($^8$C$_{11}$) 133.9 ($^8$C$_{12}$) | 3.10 (H$_3^+$, dd) 6.88 (H$_3^+$, m) 7.18 (H$_3^+$, m) | 7.30, 14.30 |
| 36 UMP | C$_4$H$_{12}$N$_2$O$_3$P | 324.18 | 105.2 (CH) 90.9 (CH) | 5.97 (CH, m) 5.98 (CH, m) | - |
| 37 Urudine | C$_4$H$_{12}$N$_2$O$_6$ | 244.20 | 72.0 (CH) 92.0 (CH) | 4.21 (CH, dd) 5.90 (CH, d) | 9.0, 4.3 |
| 38 Valine | C$_5$H$_{11}$NO$_2$ | 117.14 | 63.0 ($^3$C$_5$) 20.8 ($^3$C$_{2}$) 19.5 ($^3$C$_{4}$) | 3.60 (H$_3^+$, d) 0.98 (H$_3^+$, d) 1.03 (H$_3^+$, d) | 4.54 |
| 39 Xanthine | C$_5$H$_4$N$_2$O$_4$ | 152.11 | 140.3 (CH) | 7.90 (CH, s) | - |

Table 2. The metabolites detected in the non-polar extract of *Subramaniula asteroids*.

| Compound Name | Chemical Formula | Weight (Da) | RT (min) | Area % |
|---------------|------------------|-------------|----------|--------|
| 1 Tetrahydro-2-(12-pentadecynyloxy)-2H-pyran | C$_{20}$H$_{36}$O$_2$ | 308 | 7.20 | 1.56 |
| 2 1-Dodecanamine, N,N-dimethyl- | C$_{14}$H$_{31}$N | 213 | 11.95 | 27.96 |
| 3 2,4-Di-tert-butylphenol | C$_{14}$H$_{22}$O | 206 | 12.22 | 1.30 |
| 4 1-Tetradecanamine, N,N-dimethyl-2H-Pyran-3-ol, tetrahydro-2,2,6-trimethyl-6-(4-methyl-3-cyclohexen-1-yl)-, [3S-[3a,6a(R*)]] | C$_{16}$H$_{35}$N | 241 | 15.75 | 10.10 |
| 5 7,9-Di-tert-butyl-1-oxaspiro(4,5)decane-6,9-diene-2,8-dione | C$_{16}$H$_{36}$O$_2$ | 250 | 16.55 | 1.30 |
| 6 Phen-1,4-diol, 2,3-dimethyl-5-trifluoromethyl- | C$_{9}$H$_{9}$F$_3$O$_2$ | 206 | 18.64 | 0.92 |
| 7 Cyclopropanebutanoic acid, 2-[1-[2-[1-[2-pentylcyclopropyl]methyl cyclopropyl]methyl]cyclopropyl]methyl ester | C$_{25}$H$_{42}$O$_2$ | 374 | 19.61 | 0.91 |
| 8 n-Hexadecanoic acid (palmitic acid) | C$_{16}$H$_{32}$O$_2$ | 256 | 20.28 | 4.13 |
| 10 Hexadecanoic acid, ethyl ester | C$_{16}$H$_{38}$O$_2$ | 284 | 20.72 | 7.05 |
4. Discussion

In the current investigation, a new fungal strain of *Subramaniula asteroids* was reported. The fungus was isolated from the unique Red Sea soil habitat in Egypt. The fungus was erroneously identified as *Papulaspora* species through morphological examination [1,3]. The fungus was identified as *Subramaniula asteroids* OP484336 using molecular analysis. The fungal isolate, which was known as a member of the family Cheatomiaceae, has been recently reported as one of the new keratitis causative agents [15], and has been recorded for the first time in Egypt. Keratitis is a clinical illness that causes irritation of the eye’s cornea and is one of the primary causes of monocular blindness globally. Ocular surface disease, trauma, immunocompromised state, and contact lens usage are the major individual risk factors for keratitis [16,17]. Various microbes have the ability to cause keratitis, including viruses [18], protozoa [19], bacteria [20], and fungi [21]. Fungi keratitis contributes to 40–50% of all microbial keratitis infections [22]. However, diagnosing fungal keratitis is more difficult because a variety of fungal pathogens can cause the same disease symptoms, including *Candida*, *Fusarium*, and *Aspergillus* sp. [23]. Filamentous fungi and yeasts can both cause fungal keratitis. In tropical regions, filamentous fungal infections are prevalent [15,24]. Otherwise, yeasts are most common in countries with temperate climates [24]. One of the goals of the current study was to investigate the metabolic profile of the *Subramaniula asteroids* family, as little is known about their metabolome.

The metabolic fingerprint of the fungus generated with analytical instrumentation provides a high-precision platform for obtaining accurate information on a wide range of metabolites produced by the organism [15–24]. It can be used to give perspective into a fungal species’ biological activities, as well as play a role in taxonomic identification [6,25–28].

In the present study, NMR and GC/MS data and databases were used to determine the metabolic profile of *Subramaniula asteroids* polar and non-polar extracts. The chemical profiling of the fungal mycelium showed the presence of chemical compounds with reported biological activities. Thirty-eight metabolites were detected in the fungus polar extract based on $^1$H and $^1$H,$^13$C HSQC NMR data and NMR databases. Using combined $^1$H and $^1$H–$^13$C HSQC NMR data to examine the metabolic profile of living organisms is an accurate and efficient method [29]. The identified polar metabolites included twelve amino acids (e.g., arginine, proline, and betaine) and seven carbohydrates (e.g., trehalose, sucrose, and glucose). Amino acids and carbohydrates are important virulence factors in the pathogenicity of human pathogenic fungus [30]. The production of trehalose has been reported in various human pathogenic fungi and its accumulation has been associated with stress conditions, causing metabolism alternation and cell wall instability [31,32]. Trehalose is a critical metabolite during fungi–host infection; for example, by altering the
genes responsible for trehalose production, *Candida albicans* infection was reduced [33,34]. In addition, in *Cryptococcus neoformans*, trehalose production must be consistent in order for pathogenicity to occur [35]. The essential role of amino acids in pathogenicity of fungi has been reported [36]. Arginine, proline, and ornithine are essential for yeast-to-hyphal conversions in the human pathogen *Candida albicans* [37].

Herein, two alkaloids (oxypurinol and xanthine) have been identified in the spectral regions δ (7.5–8.5 ppm) and have been reported for further biological activities and medicinal uses. Oxypurinol has antioxidant activity [38] and it has been shown to enhance coronary and peripheral endothelial function in patients suffering from coronary artery disease [39]. Xanthine and its derivatives have been shown to have anticancer, anti-inflammatory, and antibacterial effects [40].

In the present study, twenty metabolites were identified in the non-polar extract using GC/MS. Most of these metabolites possess biological activities that have been reported in the literature. The main component was 1-Dodecanamine, and this metabolite has been reported to possess antioxidant and antimicrobial activities [41]. There were also significant amounts of 9-octadecenoic acid ethyl ester, oleic acid, and n-Hexadecanoic acid in the fungal extract. The antimicrobial activity of 9-octadecenoic acid ethyl ester against *Candida albicans*, *Bacillus subtilis*, *Pseudomonas aeruginosa*, *Escherichia coli*, and *Staphylococcus aureus* has been reported by [42]. n-Hexadecanoic acid has been shown to have anti-inflammatory, anti-rheumatic, and anticancer effects [43,44]. Oleic acid is a type of unsaturated fatty acid that has several health advantages, such as anti-cancer and anti-inflammatory characteristics, as well as effects on the immune system and skin healing [45], and showed antibacterial activity against *K. pneumonia* [5]. The present data reported a small amount of the compound Phen-1,4-diol, 2,3-dimethyl-5-trifluoromethyl: this compound exhibited antioxidant, anti-tuberculosis, and anti-clotting properties [46].

These results provide a morphological, molecular, and chemical characterization of *Subramaniula asteroids*, one of the keratitis-causing microbes. This information will be useful for learning more about the fungus and may play an important role in the fungus management of keratitis patients infected with this fungus.

5. Conclusions

A new strain of the keratitis pathogen *Subramaniula asteroids* was isolated from Hurghada, Wadi Om Nefa’a, Red Sea Governorate, Egypt. Morphological and molecular techniques were used to correctly identify the fungus. The current work is the first to provide a metabolic profile of *Subramaniula asteroids*, which can be employed in the species’ chemotaxonomical classification, antifungal drug development, and biological activity assessment.

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