Millimetre-wave spectroscopy of 2-hydroxyprop-2-enal and an astronomical search with ALMA

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ABSTRACT

Context. Several sugar-like molecules have been found in the interstellar medium (ISM). The molecule studied in this work, 2-hydroxyprop-2-enal, is among the candidates to be searched for, as it is a dehydration product of C3 sugars and contains structural motifs that are typical for some interstellar molecules. Furthermore, it has recently been predicted that it is more abundant in the ISM than its tentatively detected isomer 3-hydroxypropenal.

Aims. So far, only low-frequency microwave data of 2-hydroxyprop-2-enal have been published. The aim of this work is to deepen our knowledge about the millimetre-wave spectrum of 2-hydroxyprop-2-enal, enabling its detailed search towards astronomical objects. In particular, we target the solar-type protostar IRAS 16293–2422 and the star-forming region Sagittarius (Sgr) B2(N).

Methods. The rotational spectrum of 2-hydroxyprop-2-enal was measured and analysed in the frequency regions of 128–166 GHz and 285–329 GHz. The interstellar exploration towards IRAS 16293-2422 was based on the Atacama Large Millimeter/submillimeter Array (ALMA) data of the Protostellar Interferometric Line Survey (PILS). We also used the imaging spectral line survey ReMoCA performed with ALMA towards Sgr B2(N) to search for 2-hydroxyprop-2-enal in the ISM. We modelled the astronomical spectra under the assumption of local thermodynamic equilibrium (LTE).

Results. We provide laboratory analysis of hundreds of rotational transitions of 2-hydroxyprop-2-enal in the ground state and the lowest excited vibrational state. We report its non-detection towards IRAS 16293 B. The 2-hydroxyprop-2-enal/3-hydroxypropenal abundance ratio is estimated to be \( \lesssim 0.9–1.3 \), in agreement with the predicted value of \( \sim 1.4 \). We report the non-detection of 2-hydroxyprop-2-enal towards the hot molecular core Sgr B2(N1), and we did not detect the related aldehydes 2-hydroxypropanal and 3-hydroxypropenal either. We find that these three molecules are at least nine, four, and ten times less abundant than acetaldehyde in this source, respectively.

Conclusions. Despite the non-detections of 2-hydroxyprop-2-enal, the results of this work represent a significant improvement on previous investigations in the microwave region and meet the requirements for further searches for this molecule in the ISM.

Key words. astrochemistry – ISM: molecules – line: identification – ISM: individual objects: IRAS 16293-2422 – ISM: individual objects: Sagittarius B2 – astronomical databases: miscellaneous

1. Introduction

The discovery of the first interstellar aldehyde, formaldehyde, dates back to 1969 (Snyder et al. 1969). Since then, eight more aldehydes, including their protonated, sulphur, and cyano analogues, have been detected by astronomical observations (McGuire 2022). One of them is 2-hydroxyethanal (Hollis et al. 2000), commonly known as glycolaldehyde, a prebiotic, sugar-related molecule that fulfils the empirical formula \( \text{C}_2\text{H}_4\text{O}_2 \) (or \( \text{C}_2\text{(H}_2\text{O})_2 \)), which initiated the search for sugars in the interstellar medium (ISM). Glyceraldehyde, as the next member of the homologous series of carbohydrates, with the formula \( \text{C}_3\text{H}_6\text{O}_3 \), has not yet been successfully detected in the ISM (Hollis et al. 2004; Jiménez-Serra et al. 2020). However, its presence there was suggested by laboratory experiments with interstellar ice analogues (Layssac et al. 2020; Fedoseev et al. 2017). The same molecular formula belongs to the other sugar molecule, dihydroxyacetone (glycerone, DHA), a simple ketotriose. It was identified in the Murchison meteorite (Cooper et al. 2001), and there are reports in several studies (e.g., Apponi et al. 2006). The longer sugar with four carbon atoms, erythrulose, was another candidate sought in the ISM, although regrettably its presence there was suggested by laboratory experiments (Insausti et al. 2021). Lactaldehyde (2-hydroxypropanal, or hydroxypropionaldehyde), another oxygen-bearing member of the \( \text{C}_3 \) family of...
molecules and a methyl derivative of glycolaldehyde, has been sought, so far unsuccessfully, in three star-forming regions (Alonso et al. 2019). Nevertheless, several attempts have been undertaken to discover lactaldehyde isomers with different results. The only three C$_3$ molecules found in the ISM were ethyl formate (Belloche et al. 2009; Tercero et al. 2013), methyl acetate (Tercero et al. 2013), and hydroxyacetone (Zhou et al. 2020). Three carbon chain aldehydes discovered in the ISM are propanal (Hollis et al. 2004; Goesmann et al. 2015; Lykke et al. 2017; Yarnall et al. 2020), propenal (Dickers et al. 2001; Hollis et al. 2004; Agündez et al. 2021; Manigand et al. 2021), and propynal (Irvine et al. 1988). Propenal is presumed to be a prebiotic species not only because of its formation during the decomposition of sugars (Moldoveanu 2010; Bermúdez et al. 2013), but also because it acts as an intermediate in the prebiotic synthesis of the amino acid methionine (Van Trump & Miller 1972). Propenal also contains a vinyl group, which is a structural motif present in several interstellar molecules; one of the examples is vinyl alcohol (Turner & Apponi 2001; Agündez et al. 2021). Propenal and other aldehydes, such as propanal, were extensively studied in the (sub-/millimetre-wave region. These laboratory studies are not strictly limited to the astronomical applications but also serve to help us better understand the effects in the spectra, such as large amplitude motions or perturbations (Daly et al. 2015; Zingsheim et al. 2017, 2022).

Our molecule of interest, 2-hydroxyprop-2-enal (or 2-hydroxyacrylaldehyde, see Fig. 1), combines the structural motifs of the aforementioned vinyl alcohol and propanal. It belongs to the C$_3$H$_6$O$_2$ family of isomers. The isomerisation enthalpies for several members of this family were calculated by B3LYP/6-311++G(d,p) level of theory and basis set.

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Fig. 1. Structure of 2-hydroxyprop-2-enal displayed in the $ab$ inertial plane.

Fig. 2. Keto-enol tautomeric relationship of 2-hydroxyprop-2-enal and methylglyoxal. The most stable isomers are highlighted. The structures are accompanied by their relative zero-point corrected energy (with respect to the most stable conformer of 2-hydroxyprop-2-enal) calculated at B3LYP/6-311++G(d,p) level of theory and basis set.
the spectrum was overflowing with, of course, the unreacted precursor and by-products, such as methanol, ethanol, or trans-methylglyoxal. Trans-methylglyoxal was also observed in the original microwave study (Lovas et al. 2003).

2.2. Experiment

The spectrum of 2-hydroxyprop-2-enal was recorded with the upgraded Prague semiconductor millimetre-wave spectrometer described elsewhere (Kania et al. 2006) in two spectral ranges (128–166 GHz and 285–329 GHz). The sample pressure in the measuring glass cell varied from 15 to 30 bar. Two different glass cells with lengths of 2.3 m and 2.8 m were used, and the optical path of the cells was doubled to 4.6 m and 5.6 m, respectively, using a roof-top mirror. For the spectral measurements, we adopted the same procedure as for 2-iminopropanenitrile (Luková et al. 2022), which also revealed relatively low stability in the cell. The sample was introduced to the chamber and was kept there without pumping until the intensity of a selected rotational line decreased by 25% of its original intensity. Subsequently, the measuring cell was evacuated and filled again with the fresh sample. A frequency modulation of 28 kHz and a second harmonic lock-in detection were employed. The accuracy of the isolated measured rotational lines was 30 kHz.

3. Quantum-chemical calculations

Although the values of the rotational constants, together with the quartic centrifugal distortion constants for 2-hydroxyprop-2-enal, are known, we performed quantum-chemical calculations employing the density functional theory (DFT) combined with the B3LYP functional and Pople’s basis set with added polarisation and diffusion functions (6-311++G(d,p)). The structure optimisation, analytical calculations of second energy derivatives followed by their numerical differentiation by finite-difference methods, were performed using very tight convergence criteria and an ultra-fine grid. The harmonic and anharmonic force field calculations up to cubic level led to estimations of, for example, vibrational energies and sextic centrifugal distortion constants that supported the spectral analysis. All calculations were carried out with Gaussian 16 (Frisch et al. 2016). The four lowest-energy conformers of 2-hydroxyprop-2-enal and their spectroscopic properties are shown in Table A.1.

4. Spectroscopic results and discussion

4.1. Ground state

2-hydroxyprop-2-enal belongs to the $C_s$ symmetry point group with a Ray’s asymmetry parameter of $\kappa = -0.6$, making it a prolate asymmetric rotor. It is a planar molecule with heavy atoms located in the $ab$ inertial plane, as shown in Fig. 1. The previously determined values of electric dipole moments were $|\mu_a| = 1.146(10)$ D, $|\mu_b| = 1.560(12)$ D, and $|\mu_c| = 0$ D from symmetry (Lovas et al. 2003). Due to the high density of rotational lines originating not just from 2-hydroxyprop-2-enal, but also from the aforementioned molecules, we started the analysis in the 285 GHz region where easily identifiable groups of transition lines were expected to be observed. In series with low values of the $K_a$ quantum number, each line comprises of two $b$-type and two $a$-type transitions, as shown in Fig. 3.
Table 1. Spectroscopic constants of 2-hydroxyprop-2-enal in the ground state (G.S.) and the lowest lying excited vibrational state (A-reduction, F-representation).

| G.S. | This work | Theory | $v_{21} = 1$ |
|------|-----------|--------|-------------|
|      | Lovas et al. (2003) | This work | Theory | This work |
| $A$ (MHz) | 10 201.6867(12) | 10 201.68699(22) | 10 206.420 | 10 119.20549(88) |
| $B$ (MHz) | 4543.3353(22) | 4543.334833(81) | 4542.237 | 4541.30508(18) |
| $C$ (MHz) | 3141.7866(20) | 3141.787342(86) | 3143.335 | 3146.37761(18) |
| $\Delta_v$ (kHz) | 0.898(60) | 0.898205(72) | 0.882 | 0.91647(17) |
| $\Delta_{JK}$ (kHz) | 5.99(40) | 6.03919(20) | 6.138 | 5.99556(97) |
| $\Delta_K$ (kHz) | 5.08(25) | 5.07237(91) | 4.737 | 3.0074(33) |
| $\theta_1$ (kHz) | 0.268(15) | 0.259966(21) | 0.254 | 0.266214(22) |
| $\theta_K$ (kHz) | 4.44(94) | 4.25939(31) | 4.215 | 3.76131(84) |
| $\Phi_J$ (mHz) | 0.122(19) | 0.115 | 0.095(47) |
| $\Phi_{JK}$ (mHz) | 2.46(11) | 2.146 | -11.15(41) |
| $\Phi_{JK}$ (mHz) | -64.22(29) | -62.553 | -34.2(33) |
| $\Phi_K$ (mHz) | 105.35(93) | 103.965 | [105.35] | [103.965] |
| $\phi_1$ (mHz) | 0.0909(48) | 0.089 | [0.0909] |
| $\phi_{JK}$ (mHz) | 1.76(10) | 1.630 | 3.40(34) |
| $\phi_K$ (mHz) | 49.18(43) | 46.829 | -48.4(25) |

$J_{\text{min}}/J_{\text{max}}$, $K_{\text{min}}^v/K_{\text{max}}^v$, $N^v$, $\sigma_{\text{fit}}^v$, $\sigma_{w}^v$. Notes. (a)B3LYP6-311++G(d,p). (b)The numbers in parentheses are 1σ uncertainties (67% confidence level) in units of the last decimal digits. The SPFIT/SPCAT programme package was used for the analysis. (c)Fixed to the ground state value, which is usually a preferred constraint against the zero or poorly determined value (Urban & Sarka 1990). (d)Number of distinct frequency lines in the fit. (e)Root mean square deviation of the fit. (f)Unitless (weighted) deviation of the fit.

As this quantum number progressively increases, the lines are separated into quadruplets with higher-intensity $b$-type lines surrounding $a$-type transitions. With the initial set of transitions being fitted and predictions made, we proceeded to analyse the lower frequency part of the spectrum where we sought resembling quadruplet patterns. Apart from the $R$-branch transitions, rotational lines belonging to $Q$-branch transitions were searched and analysed as well. The search was supported by the graphical help of the Loomis-Wood-type plot from Kisiel’s Assignment and Analysis of Broadband Spectra package (Kisiel et al. 2005, 2012). Together with the data from the previous work (Lovas et al. 2003), a total set of 968 rotational transitions was employed in the final fit, out of which 594 belong to the $R$-branch with 340 $a$- and 254 $b$-type transitions, and 374 belong to the $Q$-branch $b$-type transitions. Lines in close proximity to the transitions of interfering species and lines with deformed shapes in both branches were excluded. The Pickett’s programme package SPFIT/SPCAT (Pickett 1991) with Watson’s A-reduced Hamiltonian in F representation were employed throughout the analysis. The resulting constants are summarised in Table 1.

The analysis of the ground state provides an extended set of spectroscopic constants, which is in excellent agreement with quantum-chemical calculations, as depicted in Table 1. All constants were improved by up to three orders of magnitude and a set of sextic centrifugal distortion constants has been determined for the first time.

The partition functions needed to calculate the column densities or their upper limits are provided in Table 2. The rotational partition functions for different temperatures were obtained from SPCAT covering all rotational states up to $J = 100$. The vibrational partition functions were calculated using Eq. (3.60) from Gordy & Cook (1984), employing the anharmonic frequencies of 21 vibrational modes shown in Table A.2.

4.2. Excited vibrational state $v_{21} = 1$

Our quantum-chemical calculations identified the lowest-frequency vibrational mode $v_{21}$ as the twisting skeletal vibration with the anharmonic frequency of 196.5 cm$^{-1}$. The first excited state of this mode was thus expected to be sufficiently populated at the temperature of the experiment. The predictions of its rotational constants relied on theoretical estimations of vibrational-rotational changes of the rotational constants with respect to the ground state from the first column of Table 3. Rotational transitions in $v_{21} = 1$ were displaced to higher frequencies from the ground state and the Loomis-Wood-type diagrams greatly facilitated their assignments. Only the most intense and well-distinguished lines were analysed. The dataset

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for $v_{21} = 1$ consists of 487 rotational transitions up to $K_a = 20$. The $R$-branch transitions are covered by 110 $b$-type transitions and 196 $a$-type transitions. The remaining 181 transitions belong to $b$-type $Q$-branch transitions. Not all of the visible series were added to the fit, as the signal-to-noise ratio was rather poor. The resulting spectroscopic constants are given in Table 1 while the list of measured transitions, together with those for the ground vibrational state, are listed in Table 5. The inertial defect $\Delta = -0.611 \text{ uÅ}^2$ obtained from the rotational constants is more negative than its ground-state counterpart ($\Delta = 0.083 \text{ uÅ}^2$), which confirms the out-of-plane character of the $v_{21}$ vibrational mode.

Most of the values for constants of the $v_{21} = 1$ state are in agreement with those for the ground state presented in Table 1. Yet several discrepancies occur: a difference from the values acquired by the ground-state analysis is evident for the quartic centrifugal distortion constant $\Delta K_a$ and for the $K$-dependent sextic constants. It is possible that the centrifugal distortion constants, which are dependent on this quantum number, display substantial variation with vibrational state as in acrylonitrile (Cazzoli & Kisiel 1988; Kisiel et al. 2009, 2012). However, the possible interaction with the nearby excited vibrational state at 277.4 cm$^{-1}$, or even a combination of both effects, cannot be excluded either. Unfortunately, the insufficient intensities for other low-lying vibrational states made their analysis impossible, thus hampering a further assessment of this issue. Nevertheless, the Loomis-Wood-type plots corroborate our assignment of the rotational lines.

### 5. Search for 2-hydroxyprop-2-enal towards IRAS 16293-2422

#### 5.1. Observations

The Protostellar Interferometric Line Survey is an unbiased spectral survey of the solar-type protostar IRAS 16293-2422, carried out with ALMA between 329.1 and 362.9 GHz with a spectral resolution of $\sim 0.2 \text{ km s}^{-1}$ (Jørgensen et al. 2016). The high spatial resolution of $\sim 0.5''$ clearly disentangles the emission of the two components A and B, which are separated by $\sim 5''$. The data reduction process is detailed in Jørgensen et al. (2016). Thanks to its high sensitivity (rms $\sim 4-5 \text{ mJy}$ in a bin of 1 km s$^{-1}$), a large variety of species were detected for the first time in low-mass protostars through this survey (for example, Lykke et al. 2017; Fayolle et al. 2017; Coutens et al. 2018, 2019; Manigand et al. 2020). In particular, several lines were recently assigned to 3-hydroxypropenal, HOCHCHCHO, towards component B (Coutens et al. 2022). The chemical network presented in Manigand et al. (2021) was then updated to include 3-hydroxypropenal and several of its isomers, including 2-hydroxyprop-2-enal (Coutens et al. 2022). The chemical model was in good agreement with the column density derived for 3-hydroxypropenal (within a factor of five), and with the upper limits derived for vinyl formate (C$_2$H$_5$OCHO) and 2-propenoic acid (C$_2$H$_3$COOH). The spectroscopic data were missing for the other isomers. A high abundance of 2-hydroxyprop-2-enal (6.6 $\times 10^{-4}$ with respect to CH$_3$OH) was predicted by the model, which suggested that this species could be detectable in this source.

#### 5.2. Non-detection of 2-hydroxyprop-2-enal

Thanks to the spectroscopic measurements obtained in Sect. 4, we consequently searched for 2-hydroxyprop-2-enal towards IRAS 16293 B at the exact same position where the lines of 3-hydroxypropenal were identified ($\alpha$2000 = 16$^h$32$^m$22.58$^s$, $\delta$2000 = $-24^\circ28'32.8''$). No clear detection was obtained for this species. We used the CASSIS$^1$ software for the attempted detection. We assumed a full width at half maximum (FWHM of 1 km s$^{-1}$ and a source size of 0.5'' to determine the upper limit. Complex organic molecules in this object show excitation temperatures, $T_\text{ex}$, between 125 and 300 K (Jørgensen et al. 2018). While some lines could correspond to 2-hydroxyprop-2-enal for $T_\text{ex} = 300 \text{ K}$, others are missing with the same model. The $3\sigma$ upper limit is $\sim 2.4 \times 10^{15} \text{ cm}^{-2}$ based on the undetected lines and taking into account the vibrational correction (see Table 2). For 125 K, which was slightly favoured for 3-hydroxypropenal (Coutens et al. 2022), the lines of 2-hydroxyprop-2-enal are usually blended with other species. The upper limit is, in this case, $\sim 9.4 \times 10^{14} \text{ cm}^{-2}$. The upper limit on its abundance with respect to CH$_3$OH is consequently about $\sim 2.4 \times 10^{-4}$, that is a factor of about three lower than the predictions of the chemical model, but still within the uncertainties. According to the model, 2-hydroxyprop-2-enal is mostly formed by the tautomerisation of methyl glyoxal (CH$_2$COCHO), which itself is produced essentially by the reaction between CH$_3$CO and HCO. The association of HCO and CH$_3$COH on grains, leading directly to 2-hydroxyprop-2-enal, is negligible because CH$_3$COH has a very low concentration. The derived abundance is still within the uncertainties of the chemical model, which is estimated to be a factor of 20 for this species. Indeed, the uncertainty of the branching ratios of the association reactions is estimated to be a factor of five, while an uncertainty of a factor of four comes from the destruction reactions with hydrogen atoms. As explained in Coutens et al. (2022), the tautomerisation of methyl glyoxal could be less favourable on grains than assumed in the model because this tautomerisation involves a transition state quite high in energy. The uncertainty on the production of 2-hydroxyprop-2-enal is much larger than the uncertainty on the production of methyl glyoxal (CH$_3$COCHO). The upper limit derived for the 2-hydroxyprop-2-enal/3-hydroxypropenal abundance ratio (0.9 for $T_\text{ex} = 125 \text{ K}$ and 1.3 for $T_\text{ex} = 300 \text{ K}$) is consistent with the value predicted by the chemical model ($\sim 1.4$), taking into account the uncertainties described above.

### 6. Search for 2-hydroxyprop-2-enal and related aldehydes towards Sgr B2(N1)

#### 6.1. Observations

We used astronomical data obtained with ALMA towards the high-mass star-forming protocluster Sgr B2(N). This dataset constitutes the imaging spectral line survey ReMoCA. A detailed

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1 CASSIS has been developed by IRAP-UPS/CNRS (Vastel et al. 2015). http://cassis.irap.omp.eu/
In order to search for 2-hydroxyprop-2-enal, CH\(_3\)CHO, towards Sgr B2(N1S), we relied on the LTE parameters derived for acetaldehyde, CH\(_3\)CHO, towards the same source from the ReMoCA survey, as reported in Sanz-Novo et al. (2022). These parameters are listed in Table 4. Assuming that the more complex molecule 2-hydroxyprop-2-enal traces the same region as acetaldehyde, we produced LTE synthetic spectra for the former species adopting the same parameters as for the latter, with only the column density left as a free parameter. We employed the frequency predictions derived for 2-hydroxyprop-2-enal using the spectroscopic constants from Table 1 to compute its LTE synthetic spectrum. The molecule is not detected towards Sgr B2(N1S), as illustrated in Figs. 4 and 5. The upper limit on the total column density of 2-hydroxyprop-2-enal obtained from the ReMoCA survey is reported in Table 4, after accounting for the vibrational partition function that was computed with the information provided in Sect. 4.1.

We also report in Table 4 the column density upper limits that we obtained with the ReMoCA survey for two related aldehydes: the more saturated molecule 2-hydroxypropanal, CH\(_2\)CH(OH)CHO, and 3-hydroxypropanal, HOCHCHCHO, which is a structural isomer of 2-hydroxyprop-2-enal. Neither of these two aldehydes is detected towards Sgr B2(N1S), as illustrated in Figs. B.1 and B.2, respectively. We employed the spectroscopic entry 74519 (version 1) of the Cologne Database for Molecular Spectroscopy\(^2\) (CDMS, Müller et al. 2005) to compute the LTE synthetic spectra used to derive the upper limit for the column density of 2-hydroxypropanal. This CDMS entry is based on the measurements reported in Alonso et al. (2019). The upper limit reported in Table 4 accounts for the (substantial) vibrational correction that was estimated using the anharmonic frequencies of its normal vibrational modes, computed at the same level of theory and basis set (MP2/6-311++G(d,p)) as those that Alonso et al. (2019) employed for the harmonic ones. The upper limit reported for 3-hydroxypropanal was obtained using the CDMS entry 72504 (version 1), which relies on previous microwave and sub-millimetre-wave measurements (Baughcum 1978; Stolze et al. 1981; Baba et al. 1999).

The vibrational correction was estimated using the experimental vibrational frequencies assigned by Tayyari & Milani-Nejad (1998) and measured by Smith et al. (1983). We also recall in Table 4 the upper limit obtained for propenal towards Sgr B2(N1S) from the ReMoCA survey, as reported by Sanz-Novo et al. (2022).

Table 4 shows that 2-hydroxyprop-2-enal is at least nine times less abundant than acetaldehyde towards Sgr B2(N1S). A similar limit is obtained for its structural isomer 3-hydroxypropanal, while the limits obtained for propenal and 2-hydroxypropanal are less stringent by nearly a factor of two. For comparison, the results reported in Sect. 5.2 for 2-hydroxyprop-2-enal, and those for 3-hydroxypropanal (Coutens et al. 2022)

\(^2\) [https://cdms.astro.uni-koeln.de/](https://cdms.astro.uni-koeln.de/)

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**Table 4.** Parameters of our best-fit LTE model of acetaldehyde toward Sgr B2(N1S) and upper limits for propenal, 2-hydroxypropanal, 2-hydroxyprop-2-enal, and 3-hydroxypropanal.

| Molecule                  | Status\(^{(a)}\) | \(N_{\text{det}}\)\(^{(b)}\) | \(\text{Size}^{(c)}\) | \(T_{\text{rot}}\)\(^{(d)}\) | \(\Delta V^{(g)}\)\(^{(e)}\) | \(V_{\text{off}}\)\(^{(f)}\) | \(\frac{N_{\text{det}}}{N_{\text{ref}}}\) |
|--------------------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|
| CH\(_3\)CHO\(^{(j)}\)*   | d               | 3l              | 2.0             | 250             | 6.7 (17)        | 1.09            | 5.0             | 0.0             | 1               |
| CH\(_2\)H\(_2\)CHO\(^{(j)}\) | n               | 0               | 2.0             | 250             | 1.3 (17)        | 4.46            | 5.0             | 0.0             | >5.0            |
| CH\(_3\)CH(OH)CHO, \(v = 0\) | n               | 0               | 2.0             | 250             | 1.17 (17)       | 5.68            | 5.0             | 0.0             | >4.0            |
| CH\(_2\)C\(_2\)H\(_2\)CHO, \(v = 0\) | n               | 0               | 2.0             | 250             | 7.2 (16)        | 2.41            | 5.0             | 0.0             | >9.3            |
| HOCHCHCHO, \(v = 0\)     | n               | 0               | 2.0             | 250             | 7.0 (16)        | 2.00            | 5.0             | 0.0             | >9.6            |

**Notes.** \(^{(a)}\): detection, \(^{(b)}\): nondetection. \(^{(c)}\): Number of detected lines (conservative estimate, see Sect. 3 of Belloche et al. 2016). One line of a given species may mean a group of transitions of that species that are blended together. \(^{(d)}\): Source diameter (FWHM). \(^{(e)}\): Rotational temperature. \(^{(f)}\): Total correction factor that was applied to the column density to account for the contribution of vibrationally excited states, in the cases where this contribution was not included in the partition function of the spectroscopic predictions. \(^{(g)}\): Linewidth (FWHM). \(^{(h)}\): Velocity offset with respect to the assumed systemic velocity of Sgr B2(N1S), \(V_{\text{sys}} = 62 \text{ km s}^{-1}\). \(^{(i)}\): Column density ratio, with \(N_{\text{ref}}\) the column density of the previous reference species marked with a *.

\[^{(j)}\] The parameters were derived from the ReMoCA survey by Sanz-Novo et al. (2022).
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Fig. 4. Selection of transitions of CH$_2$C(OH)CHO, $v = 0$ covered by the ReMoCA survey. The LTE synthetic spectrum used to derive the upper limit on the column density of CH$_2$C(OH)CHO, $v = 0$ is displayed in red and overlaid on the observed spectrum of Sgr B2(N1S) shown in black. The blue synthetic spectrum contains the contributions of all molecules identified in our survey so far, but does not include the contribution of the species shown in red. The central frequency is indicated in MHz below each panel as well as the half-power beam width on the left, the width of each panel in MHz in parentheses, and the continuum level of the baseline-subtracted spectra in K in brackets. The $y$-axis is labelled in brightness temperature units (K). The dotted line indicates the 3$\sigma$ noise level.

Fig. 5. Same as Fig. 4, but for CH$_2$C(OH)CHO, $v_{21} = 1$.

and acetaldehyde (Jørgensen et al. 2018), indicate that the former two molecules are $\gtrsim 130$ and $\sim 120$ times less abundant, respectively, than acetaldehyde in IRAS 16293 B. The limits obtained for both molecules in Sgr B2(N1S) are thus less constraining by more than one order of magnitude compared to IRAS 16293 B.

7. Conclusions

We present the extended dataset for the ground state of 2-hydroxyprop-2-enal, together with the first data for the low-lying excited vibrational state $v_{21} = 1$. The species was prepared in situ by the thermal decomposition of DHA and then characterised by millimetre-wave rotational spectroscopy. The effective Hamiltonian with quartic and sextic centrifugal distortion terms was employed within the analysis. The determination of sextic terms and the analysis of the $v_{21} = 1$ state were supported by quantum-chemical calculations. A total of 669 and 295 distinct frequency lines, covering 968 and 487 rotational transitions, were used in the assignment of the ground and excited vibrational states, respectively. These new laboratory measurements provided accurate frequency predictions to search for the molecule towards IRAS 16293 B and Sgr B2(N). Additionally, emission lines of the related molecules 2-hydroxypropanal and 3-hydroxypropanal were searched for in the latter source. The observational results are the following:

1. We report the non-detection of 2-hydroxyprop-2-enal towards the protostar IRAS 16293 B. The derived upper limit is in agreement, within the uncertainties, with the predictions of the chemical model developed for 3-hydroxypropanal in Coutens et al. (2022). The 2-hydroxyprop-2-enal/3-hydroxypropanal abundance ratio is estimated to be $\lesssim 0.9–1.3$, in agreement with the predicted value of $\sim 1.4$.

2. We report the non-detection of 2-hydroxyprop-2-enal towards the main hot core of Sgr B2(N) that was targeted with ALMA. We find that this molecule is at least nine times less abundant than acetaldehyde towards this source. Its column density upper limit is nearly a factor of two lower than the one derived earlier for propanal in the same source.

3. Neither of the related aldehydes, 2-hydroxypropanal and 3-hydroxypropanal, were detected towards Sgr B2(N)’s main hot core. They are at least four and ten times less abundant than acetaldehyde, respectively.

Despite the non-detections of 2-hydroxyprop-2-enal, the present work represents a significant improvement on previous laboratory studies below 25 GHz and provides an excellent base for future searches for this species in space.

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Appendix A: Supplementary tables

Table A.1 summarises the spectroscopic properties of the four lowest-energy conformers of 2-hydroxyprop-2-enal. Table A.2 lists the frequencies of the normal vibrational modes of 2-hydroxyprop-2-enal.

Table A.1. Four lowest-energy conformers of 2-hydroxyprop-2-enal.

| Conformer | I      | II     | III    | IV     |
|-----------|--------|--------|--------|--------|
| A         | MHz    | 10206  | 10038  | 9962   | 10044  |
| B         | MHz    | 4542   | 4264   | 4420   | 4239   |
| C         | MHz    | 3143   | 2993   | 3062   | 2981   |
| | $|\mu_a|$ | 1.3    | 3.33   | 4.11   | 1.3    |
| | $|\mu_b|$ | 1.65   | 0.51   | 1.29   | 1.4    |
| | $|\mu_c|$ | 0      | 0      | 0      | 0      |
| $\Delta E_{ZPE}$ | kJ/mol | 0      | 23.54  | 24.02  | 30.13  |

Table A.2. List of harmonic and anharmonic frequencies for vibrational modes of 2-hydroxyprop-2-enal calculated by B3LYP/6-311++G(d,p).

| Mode | Frequency (cm$^{-1}$) | Harmonic | Anharmonic | Symmetry |
|------|-----------------------|----------|------------|----------|
| 1    | 3680.764              | 3490.279 | A          |
| 2    | 3257.119              | 3191.542 | A'         |
| 3    | 3163.194              | 3017.537 | A          |
| 4    | 2974.271              | 2814.370 | A          |
| 5    | 1745.388              | 1712.650 | A          |
| 6    | 1707.794              | 1664.629 | A          |
| 7    | 1446.509              | 1406.382 | A          |
| 8    | 1397.543              | 1357.208 | A          |
| 9    | 1366.622              | 1340.344 | A          |
| 10   | 1255.587              | 1237.762 | A          |
| 11   | 981.869               | 964.669  | A          |
| 12   | 895.680               | 875.146  | A          |
| 13   | 681.221               | 665.809  | A          |
| 14   | 410.232               | 411.260  | A'         |
| 15   | 282.018               | 277.358  | A'         |
| 16   | 1000.470              | 990.702  | A''        |
| 17   | 891.797               | 881.615  | A''        |
| 18   | 732.165               | 716.236  | A''        |
| 19   | 558.336               | 571.796  | A''        |
| 20   | 524.626               | 471.809  | A''        |
| 21   | 201.300               | 196.519  | A''        |
Appendix B: Complementary figures:
Non-detection of 2-hydroxypropanal and 3-hydroxypropenal towards Sgr B2(N1S)

Figures B.1 and B.2 illustrate the non-detection of 2-hydroxypropanal and 3-hydroxypropenal, respectively, towards Sgr B2(N1S) with the ReMoCA survey.

Fig. B.1. Same as for Fig. 4, but for CH$_3$CH(OH)CHO, $\nu = 0$. 
Fig. B.2. Same as for Fig. 4, but for HOCHCHCHO, $v=0$. 
Fig. B.2. continued.