Analysis of the features and causes of the contamination in the Shandi River basin

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Abstract: This study examined the contamination of the surface water in the Shandi River basin in Yangquan, Shanxi province. A total of 11 sampling points (numbered S1-S2, S3-S4, S5-S11) were selected from the upstream to the downstream of the river, and 19 indexes were selected for inspection to identify the causes and features of contaminated water in the basin. The Nemerow Pollution Index was applied to evaluate the quality of the surface water and the overall extent of contamination in the basin. The result revealed that the content of Cl⁻, As, Cr⁶⁺, CN⁻, and volatile phenols met the grade III water quality standards in China, and the remaining indexes exceeded the standards by varying degrees. The five sources of contamination in the basin negatively affect the environment. The presence of contaminants at the sampling points was in the following order, from highest to lowest: SO₄²⁻ = Mn = NH₃-N > Fe²⁺ = Hg > Fe³⁺ > Cd > Zn = F > COD > NO₃⁻ = NO₂⁻. The severity of the comprehensive contamination indexes decreased in the order of S6 > S9 > S7 > S11 > S10 > S3 > S8 > S5 > S4.

Introduction
Shanxi province has abundant coal resources; however, in recent years, coal mines have been consolidated and many of them been closed down because the coal resources were exhausted after continuous mining. In Yangquan, Shanxi province, a number of goaves were abandoned because of excessive mining. After mines are closed for a period of time, mine water accumulates in the mines and pockets of water develop that become trapped in the goaves [¹], yielding acid mine drainage. As drainage continues to gather, the water level may rise and eventually overflow a mine to the surface, resulting in severe environmental damage, negative effects on people’s lives, and a high level of social and governmental concern. The Shandi River basin is a highly typical instance of this process.

The contaminated acid mine drainage within the Shandi River basin is mostly concentrated in the carbonate rock leakage sections near the Niangzi Pass spring field and greatly affects the karst underground water within the region. Liang’s [²] research stated that the springs at Niangzi Pass have been seriously contaminated because of the leakage of the acid mine drainage in parts of the region, specifically the Taohe River, Wenhe river, and Nanchuan River leakage sections, which contaminate the karst underground water. Huo’s [³] study indicated that the main routes of contamination in the Niangzi Pass spring field are the following two: Pollutants and atmospheric precipitation enter the river course and

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directly infiltrate the ground surface where carbonate rocks are exposed to supply the karst underground water; and precipitation indirectly seeps into the coverage zone of carbonate rocks and supplies the karst underground water. Mining is the most influential factor on the groundwater environment in the Niangzi Pass spring field. Wang [4] conducted an analysis and suggested that pyrite within coal seams and the surrounding strata oxidizes and hydrolyzes during coal mining. Therefore, this study focused on the surface water condition of the Shandi River and selected sampling points within the Shandi River basin on the basis of the geologic environment, mining activities, and the current condition of abandoned mines. The level, volume, and quality of the river water were monitored, and water samples were collected and analyzed to examine the actual sources and distribution of contamination along with the levels and features of the contaminants. In this study, the harm of contamination on Niangzi Pass was specified, and corresponding management measures are suggested.

1. The general situation of the studied region

The Shandi River basin is 58 km² in area, and the overall geology in this region is complex. The lithology of the stratum is relatively complete, with loess covering most of the area and bedrock partially exposed. The geological structure and the hydrogeological condition are simple. Ordovician carbonate karst aquifer is the main composition of the four types of groundwater aquifer groups that are found in the region. The aquiclude is mainly a combination of Carboniferous mudstone and sandstone, and mudstone predominates in the Upper Carboniferous Taiyuan Formation and Permian Shanxi Formation. The main source that contributes water to the aquifer is precipitation, and the water is primarily discharged from wells, springs, natural evaporation, and human activities. In 2009, the mines in the region were integrated and reorganized, and currently, only eight mines remain, with one of them being an open-pit mine.

After field studies and investigations were conducted, five sources of contamination were found to cause the contamination in the basin:

Contamination source I: Yankan coal washery. The coal washery discharges a massive amount of coal washing wastewater into the river course. Coal ash floats on the river surface, causing the river water to be black and turbid; the pH value at this sampling point has changed, showing that the river has been contaminated.

Contamination source II: Dishui Cliff comprehensive contamination source. The Yuejin mining site lies on the southwest of Dishui Cliff. The mining site is equipped with a sewage treatment plant with a treatment capacity of 10.000 m³/d. The mine water is managed before discharge; however, acid mine drainage is nevertheless observed in the mines and it flows into the Shandi River through its river course. The pH value of the sampling point was measured at 4.0, which indicates that the river has been contaminated by acid mine drainage.

Contamination source III: Xiaogou village open-pit mine contamination source. Mine water gathers in multiple mine shafts in the open-pit mine, and the sources of the gathered water are precipitation and groundwater. The gathered water is yellowish-brown and relatively turbid, and the water quality is extremely poor, with a pH value around 2.5. The gathered water infiltrates into the ground [5] directly and participates in groundwater circulation.

Contamination source IV: Shandi village Miaogou Stream contamination source. The Miaogou Stream carries surface runoff from 9# and 12# acid mine drainage. The acid mine drainage streams constantly from the outflow site and results in a flow that is reddish-brown. The tested water had a pH value of approximately 2 and was turbid with an acidic smell. When precipitation is heavy, the water volume in the ditch increases, gathers into a surface runoff, and flows into the Shandi River; accordingly, the acid mine drainage seriously contaminates the river [6].

Contamination source V: Shandi village Menlou Liu Stream. The source of the acid mine drainage at this site is uncertain, but the stream is supplied during high-flow season; therefore, the amount of leakage is higher in high-flow seasons than in low-flow seasons. The Liu Stream receives acid mine drainage from an evident outflow point near Menlou, Shandi village, and the outflow point supplies a large amount of acid mine drainage in a high-flow season and a very little in a low-flow season. A highway and a bridge are built above the stream. The stream goes through cultivated fields, flows to lower ground, and enters the river course; hence, the acid mine drainage causes serious contamination of the river.
The location of Shandi River and the distribution of mining sites and contamination sources are displayed as Figure 1.

![Figure 1. Location of Shandi River and the distribution of mining sites and contamination sources](image)

### 2. Research method

#### 2.1 Placement of sampling points

According to the hydrogeological condition of the Shandi River basin and the principle of having an even and feasible arrangement of sampling points, 11 sampling points were placed from south to north along the river. The distribution of the sampling points is depicted in Figure 2. The upstream lies to the south of Yankan, the midstream extends from Yankan to Zhainao village, and the downstream is located to the south of Zhainao village. The upstream, midstream, and downstream areas received 2, 2, and 7 sampling points, respectively, with sampling points numbered S1-S2, S3-S4, and S5-S11 in order from upstream to downstream.

![Figure 2. Distribution of the monitored sampling points in the basin](image)

#### 2.2 Selection of the analysis indexes of the water samples

The selection of the indexes for monitoring the surface water was based on the environmental quality standards for surface water and the environmental quality standard for soils in China. The main items of inspection can be categorized into two types: (1) general comprehensive indexes that reflect water quality, such as temperature, chromaticity, pH value, and electric conductivity; and (2) indexes that signify toxic substances, including the content of Pb, Cr, Cd, Hg, and organic pesticides. According to the actual situation in the Shandi River basin, this study selected a total of 19 indexes with which to assess the water samples: seven cations and anions (i.e., Fe$^{3+}$, Fe$^{2+}$, SO$_4^{2-}$, NO$_3^-$, NO$_2^-$, Cl$^-$, and F$^-$), seven heavy metal pollutants (i.e., Zn, As, Hg, Cd, Cr$^{6+}$, Pb, and Mn), and five additional indexes; CN$^-$, H$_2$S, NH$_3$-N, which cause severe water pollution, oxygen consumption demand (COD), and volatile phenols.

#### 2.3 Data analysis

ArcGIS 10.2 was employed to draft the figures depicting the distribution of sampling points and the development of each index. Excel 2007 was adopted to organize the data that reveal the content of each index and calculate the average value, over-standard rate, and the coefficient of variation (C.V). Origin 9.6 was applied to create a dotted-line chart to demonstrate the distribution of each pollution index of the sampling points. SPSS 22.0 was used to conduct a Pearson correlation analysis on each index in the water sample. Subsequently, an analysis of the contamination sources was conducted on the basis of the level of correlation.

#### 2.4 Method for evaluating the quality of surface water

The Nemerow Pollution Index method is commonly applied for evaluating the water environment of basins. This method can calculate a single pollutant index as well as the comprehensive indexes of different factors within the basin and reflect the comprehensive contamination situation [7-8].

According to the environmental quality standards for surface water (GB3838-2002) III standard value [9], the calculation formulas are as follows:

$$I_i = \frac{C_i}{C_{Oi}}$$ (1)
In formula (1) and (2), $I_i$ and $I_{i, \text{max}}$ refer to the contamination index and maximum contamination index of index $i$, respectively; $C_i$ indicates the mass concentration (mg/L) of index $i$; $CO$ represents the standard mass concentration (mg/L) of index $I$; $WQI$ stands for the comprehensive water quality indexes; and $j$ signifies the number of ions.

The degree of contamination caused by the indexes was classified into four levels to calculate a single contamination index: (1) $I_i < 1$, clean; (2) $1 \leq I_i < 2$, moderately clean; (3) $2 \leq I_i < 3$, contaminated; (4) $3 \leq I_i$, severely contaminated.

According to the results of the calculated comprehensive contamination indexes, water quality was graded into five levels ([10]): (1) $WQI < 0.80$, excellent water quality; (2) $0.81 \leq WQI < 2.50$, good water quality; (3) $2.51 \leq WQI < 4.25$, moderate water quality; (4) $4.25 \leq WQI < 7.20$, poor water quality; and (5) $WQI \geq 7.20$, very poor water quality.

3. Results and analysis

3.1 Analysis and assessments of the indexes in the water sample

3.1.1 Analysis and assessments of the cations and anions in the water sample

Table 1 presents the mass concentration of ions in the surface water at each sampling point within the Shandi River basin. The content of $\text{Fe}^{3+}$ ranged from 0 to 1.276 mg/L, the average value was 151.64 mg/L, and the over-standard rate reached 64%. $\text{Fe}^{3+}$ was assessed at sampling points S2–S11; the content of $\text{Fe}^{3+}$ at sampling points S1 and S2 was lower than the minimum detection limit of the equipment and is marked as 0. This indicated that $\text{Fe}^{3+}$ did not exceed the standard in the upstream area generally. The content of $\text{Fe}^{3+}$ reached its peak at S6, where the content of $\text{Fe}^{3+}$ severely exceeded the standard; also, $\text{Fe}^{3+}$ showed the highest degree of variation, with a C.V of 278%. The reason for this finding is that sampling point S6 is located near contamination source IV, where the acid mine drainage contamination is serious. The drainage contains abundant $\text{Fe}^{3+}$; and when the outflow increases, the content of $\text{Fe}^{3+}$ rises accordingly. Furthermore, $\text{Fe}^{3+}$ accumulates easily, and $\text{Fe}^{3+}$ content declines at a faster rate with the movement of flowing water; therefore, the spatial heterogeneity of $\text{Fe}^{3+}$ was the most apparent ([11-12]). The content of $\text{Fe}^{2+}$ ranged from 0 to 1364 mg/L, the average value was 590.86 mg/L, and the over-standard rate was 55%. $\text{Fe}^{2+}$ was evaluated at sampling points S6–S11; the content of $\text{Fe}^{2+}$ at S6-S7 and S9 was lower than the minimum detection limit of the equipment and is marked as 0. This indicated that $\text{Fe}^{2+}$ did not exceed the standard in the upstream and midstream in general. Nonetheless, the content of $\text{Fe}^{2+}$ was extremely high at S6-S7 and S9, and the reason is that S6 is located near contamination source IV and S9 adjoins contamination source V. In addition to $\text{Fe}^{3+}$, $\text{Fe}^{2+}$ is also abundant in acid mine drainage, and when the outflow increases, the content of $\text{Fe}^{2+}$ rises accordingly; in addition, $\text{Fe}^{2+}$ does not accumulate easily, so its content sampled at S7 remained high ([13]). The content of $\text{SO}_4^{2-}$ ranged from 71.40 to 19836 mg/L, the average value was 5883.09 mg/L, and the over-standard rate came to 81%. $\text{SO}_4^{2-}$ was assessed at sampling points S1–S11; the content of $\text{SO}_4^{2-}$ at S6, S7, and S9 was extremely high, which proved that the $\text{SO}_4^{2-}$ content exceeded the standard at S6, S7, and sampling points S9. The reason is that plentiful $\text{SO}_4^{2-}$ can be found in acid mine drainage, and with the abrupt increase of the outflow, the content of $\text{SO}_4^{2-}$ became exceedingly high. The content of $\text{NO}_3^-$ ranged from 0.33 to 28 mg/L, the average value was 10.22 mg/L, and the over-standard rate was 45%. $\text{NO}_3^-$ was assessed at sampling points S1–S11; the $\text{NO}_3^-$ content reached its peak at S5. The reason is that S5 is located within the range of influence of contamination source II. The secondary sewage used by the pig farm is discharged into the river course, and water quality deteriorates with the release of the highly concentrated organic sewage, which leads to the increase and exceedance of $\text{NO}_3^-$. The content of $\text{Cl}^-$ ranged from 13 to 158 mg/L, the average value was 76.59 mg/L, and the over-standard rate was 0%, which indicated that the $\text{Cl}^-$ content of the Shandi River basin met the grade III water quality standards in China. The content of $\text{F}^-$ ranged from 0.43 to 15 mg/L, the average value was 4.17 mg/L, and the over-standard rate was 45%. $\text{F}^-$ was inspected at sampling points S1–11; the content of $\text{F}^-$ was excessively high at S6, S7, and S9 and exceeded the standard because the three sampling points are near acid mine drainage contamination sources IV and V. However, acid mine drainage cannot be asserted to be the cause of the exceedance of $\text{F}^-$ because it is not a feature of acid mine drainage. Accordingly, Pearson correlation analysis was employed to analyze the $\text{F}^-$ identified in the Shandi River and the features $\text{Fe}^{3+}$, $\text{Fe}^{2+}$, and $\text{SO}_4^{2-}$ in acid mine drainage,
and the results revealed that $F^-$ has a positive correlation ($p<0.01$) with $Fe^{3+}$, $Fe^{2+}$, and $SO_4^{2-}$, and the correlation coefficients were 0.75, 0.85, and 0.98, respectively. This indicated that $F^-$ has a similar source as $Fe^{3+}$, $Fe^{2+}$, and $SO_4^{2-}$ and could further imply that contamination sources IV and V caused the exceedance of $F^-$ at S6, S7, and S9.

### 3.1.2 Analysis and Evaluation of the heavy metal factors in the water

Table 2 display the heavy metal concentration at the sampling points within the Shandi River basin. Pearson correlation analysis was employed to analyze the correlation between the heavy metals and the cations and anions in the basin to explore their possible source. The content of Zn ranged from 0.0128 to 22.4 mg/L, the average value was 5.58 mg/L, and the over-standard rate was 36%. The content of Zn inspected at sampling points S1–S11 was highest at S6 and S7. Zn had a significant correlation with the acid mine water features $Fe^{3+}$, $Fe^{2+}$, $SO_4^{2-}$, and $F^-$, which indicated that the contamination sources were similar and proved that acid mine drainage contamination source IV was the cause of the sudden increase of Zn at S6 and S7. The content of As ranged from 0.0001 to 0.05 mg/L, the average value was 0.0003 mg/L, and the over-standard rate was 0%. This demonstrated that the As content met the grade III water quality standards in China. The content of Hg ranged from 0.0002 to 0.0006 mg/L, the average value was 0.0004 mg/L, and the over-standard rate reached 40%. The content of Hg inspected at S1–S3 and in the S1 water sample was lower than the minimum detection limit of the equipment and is marked as 0. This indicated that Hg did not contaminate the river at sampling points S1–S3 and S1, but the content of Hg was relatively high at S5, S6, and S9. Hg had a significant correlation and shared a similar source with the acid mine drainage feature $Fe^{2+}$, which confirmed that the higher content of Hg at S5 was caused by acid mine drainage contamination source II, and S6–S7 were influenced by contamination sources IV and V. The content of Cd ranged from 0.0004 to 0.566 mg/L, the average value was 0.1283 mg/L, and the over-standard rate reached 55%. Sites S6, S7, and S9 were inspected and found to have a higher content of Cd than sampling points S1–11.

**Table 1.** Mass concentration and the distributions of ions in the surface water collected at the sampling points in the Shandi River basin

| Items     | Fe$^{3+}$ (mg·L$^{-1}$) | Fe$^{2+}$ (mg·L$^{-1}$) | $SO_4^{2-}$ (mg·L$^{-1}$) | NO$^+$ (mg·L$^{-1}$) | NO$^-$ (mg·L$^{-1}$) | Cl$^-$ (mg·L$^{-1}$) | F$^-$ (mg·L$^{-1}$) |
|-----------|-------------------------|-------------------------|---------------------------|----------------------|----------------------|----------------------|---------------------|
| Minimum   | 0                       | 0                       | 71.40                     | 0.33                 | 0.02                 | 13                   | 0.43                |
| Maximum   | 1276                    | 1364                    | 19836                     | 28                   | 34.6                 | 158                  | 15                  |
| Average   | 151.64                  | 590.86                  | 5883.09                   | 10.22                | 5.63                 | 76.59                | 4.17                |
| Standard  | 0.30                    | 0.30                    | 250                       | 10                   | 10                   | 250                  | 1                   |
| Over-standard rate | 64                    | 55                       | 81                       | 45                   | 9                    | 0                    | 45                  |
| C.V       | 2.78                    | 1.05                    | 1.32                      | 0.94                 | 2.20                 | 0.64                 | 1.35                |

3.1.3 Analysis and evaluation of the content of $CN^-$, volatile phenols, $H_2S$, $NH_3N$, and COD in the water sample

Table 3 display the mass concentration of $CN^-$, volatile phenols, $H_2S$, $NH_3N$, and COD at the sampling points placed
within the Shandi River basin. Pearson correlation analysis was employed to analyze the relationships between the cations and anions and the content of CN\(^-\), volatile phenols, H\(_2\)S, NH\(_3\)-N, and COD separately to explore their possible sources, and the results are listed in Table 4. The content of CN\(^-\) was lower than the minimum detection limit of the equipment at sampling points S1–S11, which indicated that the CN\(^-\) content met the grade III water quality standards in China. The content of volatile phenols ranged from 0.002 to 0.005 mg/L, the average value was 0.0033 mg/L, and the over-standard rate was 0%. This also indicated that the content of volatile phenols within the Shandi River basin met the grade III water quality standards in China. The content of H\(_2\)S ranged from 0 to 0.35 mg/L, the average value was 0.1045 mg/L, and the over-standard rate was 18%. The H\(_2\)S content inspected at S1, S2, S9, and S11 was relatively high. According to Table 4, H\(_2\)S had no significant correlation with the acid mine drainage ions, proving that their contamination sources were different and the reason for the exceedance remains to be identified. The content of NH\(_3\)-N ranged from 6.22 to 63.8 mg/L, the average value was 22.0333 mg/L, and the over-standard rate reached 81%. The content of NH\(_3\)-N inspected at S3, S6, S7, and S9 was relatively high. Based on Table 4, NH\(_3\)-N had a positive correlation with the acid mine drainage feature ions Fe\(^{2+}\) and SO\(_4^{2-}\). This indicated that the contamination sources were similar and confirmed that mine washery contamination source I was the cause of the higher NH\(_3\)-N content at S3, whereas acid mine drainage contamination sources IV and V were the reason for the higher NH\(_3\)-N content at S6, S7, and S9. The content of NH\(_3\)-N ranged from 6.22 to 63.8 mg/L, and the over-standard rate reached 81%. The content of COD ranged from 0.8 to 240 mg/L, the average value was 49.3785 mg/L, and the over-standard rate was 27%. The concentration of COD inspected at S7 and S9 was relatively high. Based on Table 4, the COD concentration has no significant correlation with the acid mine drainage features; accordingly, the contamination sources were different, and the reason for the contamination remains to be known.

Table 3. Mass concentration and the distributions of the factors collected at the sampling points in the Shandi River basin

| Items                        | CN$^-$ | Volatile phenols | H$_2$S | NH$_3$-N | COD |
|------------------------------|--------|-----------------|--------|----------|-----|
| Minimum value/mg·L$^{-1}$    | 0      | 0.002           | 0      | 6.22     | 0.8 |
| Maximum value/mg·L$^{-1}$    | 0      | 0.005           | 0.35   | 63.8     | 240 |
| Average value/mg·L$^{-1}$    | 0      | 0.0033          | 0.1045 | 22.0333  | 49.3785 |
| Standard value/mg·L$^{-1}$   | 0.02   | 0.005           | 0.2    | 1        | 20  |
| Over-standard value/%        | 0      | 0               | 18     | 81       | 27  |
| C.V                         | -      | 45%             | 126%   | 89%      | 173%|

Table 4. Correlation analysis between the factors CN$^-$, volatile phenols, H$_2$S, NH$_3$-N, and COD and ions

| CN$^-$ | Phenol | H$_2$S | NH$_3$-N | COD | Fe$^{2+}$ | Fe$^{3+}$ | SO$_4^{2-}$ | NO$_3^-$ | NO$_2^-$ | Cl$^-$ | F   |
|--------|--------|--------|----------|-----|-----------|-----------|------------|----------|----------|--------|-----|
| CN$^-$ | 1      |        |          |     |           |           |            |          |          |        |     |
| Phenol | -      | 1      |          |     |           |           |            |          |          |        |     |
| H$_2$S | -0.391 |        | 1        |     |           |           |            |          |          |        |     |
| NH$_3$-N | -0.11 | -0.108 | 1        |     |           |           |            |          |          |        |     |
| COD    | -0.124 | 0.530  | 0.358    | 1   | -0.111    | 1         |            |          |          |        |     |
| Fe$^{2+}$ | -0.244 | -0.129 | -0.788** | -0.111 | 1         |           |            |          |          |        |     |
| Fe$^{3+}$ | -0.249 | 0.156  | 0.852**  | 0.599 | 0.673**   | 1         |            |          |          |        |     |
| SO$_4^{2-}$ | -0.169 | -0.123 | 0.813**  | 0.356 | 0.694*    | 0.928**   | 1          |          |          |        |     |
| NO$_3^-$ | -0.103 | -0.367 | -0.474   | -0.462 | -0.339    | -0.645*   | -0.608*    | 1        |          |        |     |
| NO$_2^-$ | 0.582  | -0.266 | 0.230    | -0.061 | -0.119    | -0.217    | -0.220     | 0.188    | 1        |        |     |
The results demonstrated that the concentration of Fe, Mn, Cd, and Hg significantly increased near the contamination source I; stated differently, contamination source I had a greater influence on the content of NO$_3^-$ and NH$_3$-N in the river. The concentrations of NO$_3^-$ and Hg significantly increased near contamination source II; specifically, contamination source II had a greater influence on the content of NO$_3^-$ and Hg in the river. The concentration of NO$_3^-$ and Hg somewhat intensified near contamination source III, which would suggest that contamination source III had a greater influence on the content of NO$_3^-$ and Hg in the river. The concentrations of Fe$^{3+}$, Fe$^{2+}$, SO$_4^{2-}$, F, Zn, Hg, Cd, and Mn were significantly elevated near contamination source IV; therefore, contamination source IV most strongly influenced the content of Fe$^{3+}$, Fe$^{2+}$, SO$_4^{2-}$, F, Zn, Hg, Cd, and Mn in the river. The concentrations of Fe$^{2+}$, SO$_4^{2-}$, F, Hg, Cd, and Mn were significantly higher at sampling sites near contamination source V; accordingly, contamination source V can be said to have most strongly affected the content of Fe$^{2+}$, SO$_4^{2-}$, F, Hg, Cd, and Mn in the river. According to the results, the five contamination sources all negatively affected the environment of the basin, particularly acid mine drainage contamination sources IV and V. The mass concentration of most indexes fluctuated within a wide range, but the patterns of the fluctuation were identical: values increased abruptly near the contamination source but generally decreased in the order of downstream >midstream >upstream.

### 3.1.4 Summary

Based on the preceding analysis results, the content of Cl$^-$, As, Cr$^{6+}$, Pb, CN$^-$, and volatile phenols met the grade III water quality standards in China, whereas the remaining indexes exceeded the standards by varying degrees. The concentrations of NO$_3^-$ and NH$_3$-N were higher near contamination source I; stated differently, contamination source I had a greater influence on the content of NO$_3^-$ and NH$_3$-N. The concentrations of NO$_3^-$ and Hg significantly increased near contamination source II; specifically, contamination source II had a greater influence on the content of NO$_3^-$ and Hg in the river. The concentration of NO$_3^-$ and Hg somewhat intensified near contamination source III, which would suggest that contamination source III had a greater influence on the content of NO$_3^-$ and Hg in the river. The concentrations of Fe$^{3+}$, Fe$^{2+}$, SO$_4^{2-}$, F, Zn, Hg, Cd, and Mn were significantly elevated near contamination source IV; therefore, contamination source IV most strongly influenced the content of Fe$^{3+}$, Fe$^{2+}$, SO$_4^{2-}$, F, Zn, Hg, Cd, and Mn in the river. The concentrations of Fe$^{2+}$, SO$_4^{2-}$, F, Hg, Cd, and Mn were significantly higher at sampling sites near contamination source V; accordingly, contamination source V can be said to have most strongly affected the content of Fe$^{2+}$, SO$_4^{2-}$, F, Hg, Cd, and Mn in the river. According to the results, the five contamination sources all negatively affected the environment of the basin, particularly acid mine drainage contamination sources IV and V. The mass concentration of most indexes fluctuated within a wide range, but the patterns of the fluctuation were identical: values increased abruptly near the contamination source but generally decreased in the order of downstream >midstream >upstream.

### 3.2 Water quality evaluation analysis and results

#### 3.2.1 Analysis and results of each single contamination index

The evaluation of the influence level of each single contamination index was calculated using single contamination indexes. The results demonstrated that the river was not contaminated by Cl$^-$, As, Cr$^{6+}$, Pb, CN$^-$, volatile phenols, or H$_2$S, but it was contaminated by Fe$^{3+}$, Fe$^{2+}$, SO$_4^{2-}$, NO$_3^-$, NO$_2^-$, F, Zn, Hg, Mn, Cd, NH$_3$N, and COD by varying degrees. The severity of each single contamination index descends in the following listed order: Fe$^{2+}$ > Fe$^{3+}$ > Mn > NO$_3^-$ > Cd > NH$_3$-N > Zn > F > Hg > COD > NO$_3^-$ > NO$_2^-$. The proportion of each contamination index at the sampling points is in the order of SO$_4^{2-}$ = Mn = NH$_3$-N > Fe$^{2+}$ = Hg > Fe$^{3+}$ > Cd > Zn = F > COD > NO$_3^-$ = NO$_2^-$. This indicated that the content of Fe$^{2+}$ caused the highest level of contamination, and the content of SO$_4^{2-}$, Mn, and NH$_3$-N had the widest contamination range. The distribution of each contamination index was displayed in Figure 3. Fe$^{3+}$, Fe$^{2+}$, SO$_4^{2-}$, F, Zn, Hg, Mn, Cd, and NH$_3$N had relatively high contamination indexes near sampling point S6. The contamination source analysis showed that S6 was influenced by contamination source IV, which resulted in the high contamination indexes of the factors listed above. The content of Fe$^{2+}$, SO$_4^{2-}$, F, Zn, Hg, Mn, Cd, NH$_3$N, and COD also exhibited increases. The contamination source analysis revealed that S9 was affected by contamination source V, which resulted in high contamination indexes in the factors Fe$^{3+}$, SO$_4^{2-}$, F, Zn, Hg, Mn, Cd, and NH$_3$N, whereas the contamination source of COD remains to be clarified. The outflow of acid mine drainage led to a critical level of contamination at S6 and S9.
3.2.2 Results and analysis of the comprehensive contamination indexes

The evaluation results concerning Shandi River water quality comprehensive contamination indexes were obtained through the calculation of the heavy metal comprehensive indexes at the sampling points (see Table 5). The results revealed that only the water quality at S1 and S2 was graded excellent, whereas that at the other nine sampling points was very poor. The severity of the comprehensive contamination indexes decreased in the order of S6 > S9 > S7 > S11 > S10 > S3 > S8 > S5 > S4. Areas with severe and long-term contamination surrounded the Shandi River village, and the two areas with acid mine drainage contamination were in the most impaired condition.

Table 5. Result of the evaluation of Shandi River water quality

| ID | WQI   | Grading results of the water quality |
|----|-------|-------------------------------------|
| S1 | 0.74  | Excellent                           |
| S2 | 0.42  | Excellent                           |
| S3 | 265.69| Very poor                           |
| S4 | 16.97 | Very poor                           |
| S5 | 39.69 | Very poor                           |
| S6 | 516804.44 | Very poor                        |
| S7 | 2958400.00 | Very poor                        |
| S8 | 98.01 | Very poor                           |
| S9 | 3062500.00 | Very poor                        |
| S10| 22151.36 | Very poor                         |
| S11| 1951166.92 | Very poor                         |

4. Conclusion and suggestions

This study conducted field surveys and discovered the distribution of the contamination sources in the Shandi River basin; based on the results of the surveys, sampling points were placed at the contamination sources to obtain water sample data for water quality evaluation. The results are as follows:

1) The content of Cl-, As, Cr6+, CN-, and volatile phenols met the grade III water quality standards, whereas the remaining indexes all exceeded the standards by varying degrees. Five contamination sources were found within the basin, and each source had a negative influence on the water quality in the basin. Contamination source I had a major influence on the content of NO2- and NH3-H in the river; contamination source II caused the high content of NO3- and Hg; contamination source III affected the content of NO3- and Hg; contamination source VI increased the content of Fe3+,
Fe^{2+}, SO_{4}^{2-}, F, Zn, Hg, and Mn; and contamination source V was the cause of elevated levels of Fe^{2+}, SO_{4}^{2-}, F, Hg, and Mn in the river. The patterns of the variations in mass concentration were identical in terms of most factors: values increased abruptly near the contamination sources and generally decreased in the order of downstream > midstream > upstream.

(2) The river was not contaminated by the content of Cl, As, Cr, Pb, Br, CN, volatile phenols, or H2S but was contaminated by Fe^{2+}, Fe^{3+}, SO_{4}^{2-}, NO_{3}^{-}, NO_{2}^{-}, F, Zn, Hg, Mn, Cd, NH_{3}.N, and COD by varying degrees. The level of contamination by each index decreased in the order of Fe^{2+} > Fe^{3+} > Mn > SO_{4}^{2-} > Cd > NH_{3}.N > Zn > F > Hg > COD > NO_{3}^{-} > NO_{2}^{-}. Fe^{2+} was responsible for the highest level of contamination within the Shandi River basin, and SO_{4}^{2-}, Mn, and NH_{3}.N exhibited the widest range of contamination.

(3) Within the Shandi River basin, only the water quality at S1 and S2 was graded excellent, whereas that at the remaining sampling points was very poor. The severity of the comprehensive contamination indexes decreased in the order of S6 > S9 > S7 > S11 > S10 > S3 > S8 > S5 > S4. Areas with serious and long-term contamination exist around the Shandi River village, and the two areas with acid mine drainage contamination were in the most severe condition.

Given the contamination within the Shandi River basin, upstream regulations are suggested to lower the harm to the Niangzi Pass spring field. Abandoned open-pit mines should be filled, and the garbage in the mines should be disposed of properly. The mine shafts should be sealed to prevent the oxidation of mine water. Additionally, on-site processing measures are recommended (e.g., permeable reactive barriers (PRB) or phytoremediation techniques) to deal with the contamination that has already been caused by acid mine drainage.

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