In silico investigation of ACE2 and the main protease of SARS-CoV-2 with phytochemicals from Myristica fragrans (Houtt.) for the discovery of a novel COVID-19 drug

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Abstract
Severe acute respiratory syndrome coronavirus 2 (SARS-CoV-2), the virus that causes coronavirus disease 2019 (COVID-19), is a new coronavirus strain that was first reported in December 2019 in Wuhan, China. A specific treatment for COVID-19 has yet to be identified. Potential therapeutic targets include SARS-CoV-2 main protease (Mpro) and the SARS-CoV-2 spike-ACE2 interaction. Molecular docking, molecular dynamics (MD), solvent screening for the extraction of the specified compounds, and prediction of the drug properties of certain molecules were the methods used in this study to investigate compounds from the medicinal plant Myristica fragrans, which is one of twelve herbs in Prasachandaeng remedy (PSD). ArgusLab, AutoDock Vina, and AutoDock were used to perform docking tasks. The examined ligands were compared with panduratin A as a standard (Kanjanasirirat et al., 2020), which is a promising medicinal plant molecule for the treatment of COVID-19. Molecular docking revealed that malabaricones B and C and licarins A, B and C bound to SARS-CoV-2/ACE2 and SARS-CoV-2 Mpro with low binding energies compared to that of the standard ligand. Furthermore, appropriate solvent usage is important. Acetone was selected by COSMOquick software for compound extraction in this investigation because it can extract large amounts of all five of the abovementioned M. fragrans compounds. Furthermore, the drug-like properties of these compounds were studied utilizing the Lipinski, Veber, and Ghose criteria. The results revealed that these M. fragrans compounds have potential as effective medicines to combat the COVID-19 pandemic. However, to assess the therapeutic potential of these ligands, additional research is needed, which will use our findings as a foundation.

1. Introduction

The coronavirus (CoV) family includes severe acute respiratory syndrome coronavirus 2 (SARS-CoV-2), which was first discovered in Wuhan, China. Coronavirus disease 2019 (COVID-19) is a highly contagious human virus that has spread rapidly and caused an unprecedented pandemic, resulting in a large number of deaths and economic crises around the world (Prajapat et al., 2020, Sohrabi et al., 2020). As of December 14, 2021, the World Health Organization (WHO) reported that over 269.5 million cases of COVID-19 had been confirmed, with 5,304,248 deaths (World Health Organization, 2021). People in developing countries have been the most affected in terms of public health because they lack access to modern health care systems and treatments (Dandara et al., 2021). However, phytotherapy, which uses natural compounds to treat viral infections, could be a viable option (Antonio et al., 2021). Traditional medicines, primarily medicinal herbs,
are used by approximately 80% of the population in underdeveloped nations, according to WHO estimates (Series, 2018). Random selection of natural sources, chemotaxonomy (screening of comparable compounds in organisms belonging to the same family or genus), and ethnopharmacology are the three major approaches used to choose natural products for the study of their biological properties (Sheng-Ji, 2001, Singh et al., 2020). The most effective strategy has been determined to be the study of natural products that have a long history of use in diverse communities for the treatment of certain diseases and are part of the phytotherapeutic arsenal of common knowledge (Wu and Tan, 2019).

Prasachandaeng remedy (PSD), which is on Thailand’s National List of Essential Treatment, has been used as an antipyretic medicine for children and adults in Thailand’s hospitals and TTM clinics (Prommee et al., 2021). PSD is made up of twelve herbs, i.e., Citrus aurantifolia (Christm.) Swingle (Ma-Nao), Draeena cochinchenensis (Lour.) S.C. Chen (Chan-Daeng), Bouea macrophylla Griff. (Ma-Prang), Ligusticum chaunxiong Hort. (Khod-Hua-Bua), Caesalpinia sappan L. (Faang), Kaempferia galanga L. (Por-Hom), Mesua ferrea L. (Boon-Nak), Mammea siamensis T. Anderson. (Sa-ra-Pee), Jasminum sambac (L.) Alton (Ma-li), Heliocissus terminalis (Kurz) Sleumer (Mhuad-Kon), Nelumbo nucifera Gaertn (Bua-Luang), and Myristica fragrans Houtt. (Chan-Thet). Although PSD contains twelve herbs, the M. fragrans extract was determined to be the most efficient against COVID-19; however, there is no scientific evidence that it inhibits COVID-19. As a result, the current study focused on phytocomponents from M. fragrans Houtt., which are indicated in Table 1 as COVID-19 inhibitors.

M. fragrans (Houtt.) is an aromatic evergreen tree with spreading branches and golden fleshy fruits that resemble apricots or peaches to grow at a height of 30–39 feet (1 ft = 30.48 cm). Widely known as nutmeg, it is one of twelve herbs in PSD. There are several additional species of this plant, but M. fragrans is the most common. Both nutmeg and mace are used to flavor a variety of meals, including breakfast cereals, snacks, soups, sweet sauces, fruit juices, and gravies (Hallström and Thuvander, 1997). Because of its essential oil, M. fragrans has been used to treat rheumatoid arthritis and gastrointestinal ailments in China, Malaysia, and Indonesia. Additionally, the administration of M. fragrans to hyperlipidaemic rabbits was shown to lower blood lipoprotein lipid levels (Ram et al., 1996).

Many studies have shown that active substances with antiviral effects can be extracted from M. fragrans, such as two neolignan compounds and the well-recognized as resorcinols malabaricone A, B, and malabaricone C, which also possess known antiviral activities and the well-recognized as resorcinols malabaricone effects can be extracted from lipid levels (Ram et al., 1996).


tag{panduratin A} isoelemicin myristicin

| Compounds     | ArgusLab  | Autodock | Autodock Vina | Inhibitory activity IC_{50} |
|---------------|-----------|----------|---------------|----------------------------|
|               | Mpro      | Mpro     | Mpro          | Mpro                       |                           |

Table 1 Molecular docking results and predicted IC_{50} values of SARS-CoV-2/ACE-2 and SARS-CoV-2 main protease inhibitory activity of 13 Myristica fragrans (Houtt.) extracts and Panduratin A (positive control).

Our study relies on an in silico approach similar to that used in the early stages of today’s state-of-the-art drug discovery pipelines, which includes (1) protein–ligand docking of all M. fragrans bioactive compounds with the binding interfaces of the SARS–CoV-2 main protease Mpro and the receptor binding domain (RBD)–ACE2 complex; (2) ligand pathway simulations of the best predicted compounds from step 1 to evaluate convenient mechanisms by which the compounds can enter the binding site; (3) molecular dynamics (MD) simulations to determine the stability of the best fit protein–ligand complexes from step 1; (4) solvent evaluation to determine the maximum yields of the selected compounds; and (5) calculation of the pharmacokinetic parameters for the most qualified compounds based on the docking protocol’s preceding sections. This study shows that M. fragrans-based compounds have antiviral activity and can be used as quick phytotherapeutic alternatives for treating COVID-19.

2. Materials and methods

2.1. Materials

In this investigation, the following software programs were used: i) AutoDock 1.5.6, ii) Python 3.8.2, iii) MGLTools 1.5.4, iv) Discovery Studio 2017, v) Ligplot, vi) ArgusLab 4.0.1, vii) ChemSketch, viii) Avogadro, ix) OpenBabel, and x) COSMOquick for the screening the appropriate extraction solvents; xi) SwissADME (a free pharmacokinetics evaluation web tool), and xii) PlayMolecule (a free MD online tool). CavityPlus is a web server that can detect protein cavities. The current investigation was conducted with a system with the following properties: processor, Intel Xeon-E5-2678v3 12C/24 T CPU @ 2.50 GHz – 3.10 GHz; system memory, 32 GB RAM DDR4-2133 RECC; graphics processor, VGA GTX 1070 Ti 8G; operating system type, 64-bit; and operating system, Windows 10. These prerequisites were specified in the documentation of the abovementioned software.

2.2. Protein structure and setup

The calculated binding affinities of the main constituents of M. fragrans (Table 1) and their capacity to destroy the SARS-CoV-2/ACE2 complex and inhibit SARS-CoV-2 Mpro, a simple therapeutic strategy for anti-coronavirus therapy, were investigated. The crys-
tal structures of the SARS-CoV-2/ACE2 complex and SARS-CoV-2 Mpro were retrieved from the Protein Data Bank (PDB entry codes 7DMU and 7KVL, respectively) (Higuchi et al., 2021; Noske et al., 2021), and Discovery Studio 2017 was used to manually remove all bound ligands, ions, and solvent molecules. AutoDock Tools was used to parameterize the structures of the chosen proteins for docking investigations (Eawsakul et al., 2021; Nasongkla et al., 2021). Polar hydrogens were added to aid in the creation of hydrogen bonds.

2.3. Molecular docking

ArgusLab was used to dock 13 compounds derived from M. fragrans into the target sites of SARS-CoV-2/ACE2 and SARS-CoV-2 Mpro. The binding values were verified using AutoDock Vina, and the binding energies were calculated using the following parameters: the grid box with xyz points was set to a size of $66 \times 66 \times 80$ with grid positions of 39.905, 3.092, and 22.477 and an exhaustiveness number of 20. The xyz points of the SARS-CoV-2 major protease binding pocket were set to $52 \times 70 \times 64$ with grid positions of $-26.283, 12.599,$ and $58.966$ and an exhaustiveness number of 20. Furthermore, AutoDock was utilized to confirm the binding affinities using the same box parameters as AutoDock Vina. The Lamarckian genetic algorithm (LGA) was used to find the probable docking conformations of the ligands in the targeted proteins, with the number of GA iterations set to 50. Default values were used for all other parameters (Eawsakul et al., 2021; Nasongkla et al., 2021). The results are expressed in terms of binding energy (kcal/mol), and Eq. (1) was used to predict the 50% inhibition concentration ($IC_{50}$). Docking effects were considered when the binding energy values were less than those of panduratin A to the target proteins in at least 1 program.

\[
\text{Inhibition constant (Ki) } = \frac{IC_{50}}{1 + L/Kd}. \tag{1}
\]

where $K_i$ is the inhibition constant, which was calculated from $K_i = \exp(\Delta G/R \times T)$; $K_d$ is the dissociation constant; and $L$ is the ligand concentration.

Discovery Studio and Ligplot were used to display the chemical interactions (hydrogen bonds and hydrophobic interactions) between the binding ligand and the site on the target protein.

2.4. Analysis of the active pocket binding site

CavityPlus software (Xu et al., 2018) was used to assess the binding location of the ligand–protein combination (pkumdl.cn). Druggability was quantitatively calculated using the binding sites of the entire protein. The druggability value indicates whether a cavity is a good target for binding M. fragrans compounds.

2.5. MD simulations

Starting with the output model from the docking experiments, the ligand–protein complex simulations were carried out using the AMBER force field (Gnanocelli et al., 2020) and the parametrization function. Using pH = 7.4, the AMBER force field, and the default experimental parameters, the complexes were built for simulation using the ProteinPrepare and SystemBuilder modules (Coghi et al., 2021). SimpleRun was used to perform a 25 ns simulation using the default settings (Gnanocelli et al., 2020; Coghi et al., 2021). The outputs included the root mean square deviation (RMSD) for each frame against time and the root mean square fluctuation (RMSF) of the protein.

2.6. Solvent considerations for selected compounds

Following our previous study (Eawsakul et al., 2021), COSMOquick was used to determine which solvents were suitable for extraction to improve the extraction yields of the selected compounds (Achini et al., 2021). Five compounds were evaluated in nine different solvents, including water, ethanone, methanol, n-butanol, dichloromethane (DCM), chloroform, acetone, n-hexane, and diethyl ether. Physical and chemical criteria, such as solubility, dielectric constant, and donor and acceptor properties, were used to classify the solvents. Four factors were evaluated in this investigation for the selected compounds: heat of fusion ($H_{fus}$), temperature ($T$; set to room temperature), and the melting temperature ($T_m$) of the compounds. Thus, using the COSMOquick mathematical software technique and Eq. (2), the solubilities of the selected compounds in various solvents were computationally predicted.

\[
\Delta G_{fus} = \Delta H_{fus} - \Delta H_{fus} (1 - T)/T_m. \tag{2}
\]

2.7. Drug-likeness and absorption, distribution, metabolism, excretion, and toxicity (ADMET) prediction for M. Fragrans compounds

Computer-based ADME evaluations have recently gained popularity in drug discovery and development (Ntie-Kang et al., 2013). The SwissADME server (https://www.swissadme.ch/index.php) was used to estimate the pharmacokinetic properties and drug-likeness of the drug candidate molecules (Ranjith et al., 2019). Additionally, the Lipinski, Ghose, and Veber bioavailability criteria and scores were also applied for toxicological predictions (Kadri and Aouadi 2020).

3. Results

3.1. Molecular docking

Plant chemicals have long attracted scientific interest in the development of new drugs. Despite the use of various antiviral medications to treat COVID-19, experimental and clinical therapeutic development investigations still take a long time. In this study, molecular docking revealed robust interactions between certain compounds and SARS-CoV-2/ACE2 and SARS-CoV-2 Mpro. The scoring function based on the AScore and LGA was used to determine the binding strength. Electrostatic, hydrogen bonding, and van der Waals interactions may all contribute to the binding free energy (Mihajlovic and Mitrasinovic 2009). The protein–ligand complex with the lowest binding energy reflects the greatest stability. Table 1 shows the binding energy results, which were determined by ArgusLab, AutoDock, and AutoDock Vina. Discovery Studio 2017 (Bovia 2017) and Ligplot were used to visualize all of the docked structures (Figs. 1-12). Panduratin A, a positive control, binds with high affinity to SARS-CoV-2/ACE2 and SARS-CoV-2 Mpro (-13.9 and -10.7 kcal/mol, respectively, as determined by ArgusLab software; -7.2 and -7.0, respectively, as determined by AutoDock software; and -9.3 and -6.6, respectively, as determined by AutoDock Vina software).

3.2. MD simulations

Since molecular docking studies of compound candidates binding to target proteins may not be sufficient, the use of MD is very important. MD is performed to investigate the stability of ligand–protein complexes under a system designed to approximate the conditions in the human body. In this study, stability and flexibility were determined by calculating the RMSD and RMSF values,
Fig. 1. Binding position of licarin A (purple), licarin B (yellow), licarin C (red), and positive control as panduratin A (green) on ACE2.

Fig. 2. 3D (A), 2D (B) interaction visualization of panduratinA binding ACE2 with discovery studio and 2D (C) visualization with Ligplot.

Fig. 3. 3D (A), 2D (B) interaction visualization of licarin A binding ACE2 with discovery studio and 2D (C) visualization with Ligplot.

Fig. 4. 3D (A), 2D (B) interaction visualization of licarin B binding ACE2 with discovery studio and 2D (C) visualization with Ligplot.
respectively. During simulations, high deviations and protein fluctuation can indicate a lack of stability.

According to the MD results, the RMSD values of ACE2 with licarins A, B, and C seem to be stable compared to those of ACE2 without a ligand. The initial RMSDs after the first 10 ns and the final RMSDs for SARS-CoV-2/ACE2 in complex with panduratin A as a positive control were 1.537 and 2.026 nm, respectively, over the entire simulation period (range: 0.489 nm). Thus, good chemical candidates should have an RMSD range of less than 0.489 nm. During the simulation period, the initial RMSD after the first 10 ns and the final RMSDs for the ACE2 protein-licarin A complex were 1.445 and 1.681 nm, respectively (range: 0.236 nm). The corre-
Fig. 8. 3D (A), 2D (B) interaction visualization of licarin A binding Mpro with discovery studio and 2D (C) visualization with Ligplot.

Fig. 9. 3D (A), 2D (B) interaction visualization of licarin B binding Mpro with discovery studio and 2D (C) visualization with Ligplot.

Fig. 10. 3D (A), 2D (B) interaction visualization of licarin C binding Mpro with discovery studio and 2D (C) visualization with Ligplot.

Fig. 11. 3D (A), 2D (B) interaction visualization of malabaricone B binding Mpro with discovery studio and 2D (C) visualization with Ligplot.
The corresponding values for the ACE2 protein-licarin B complexes were 1.956 and 1.907 nm, respectively (range: 0.049 nm), and those of the ACE2 protein-licarin C complex were 1.718 and 1.694 nm, respectively (range: 0.024 nm). Thus, the RMSD values of licarins A, B, and C were smaller than that of the positive control.

The residual atom fluctuations of the ACE2-panduratin A complex (positive control) were found to be in the range of 0.56–8.95 nm (fluctuation: 8.39 nm), as illustrated in Fig. 13B. On the other hand, licarins A, B, and C had ranges of 0.49–6.02 nm (fluctuation: 5.53 nm), 0.51–7.85 (fluctuation: 7.34 nm), and 0.50–6.36 nm (fluctuation: 5.85 nm), respectively. The residual atom fluctuations of Mpro with panduratin A were found to lie in the range of 0.47–9.52 nm (fluctuation: 9.05 nm), as illustrated in Fig. 14B. In contrast, licarin A, licarin C, malabaricone B and malabaricone C in complex with Mpro were found to have residual atom fluctuation ranges of 0.45–7.71 nm (fluctuation: 7.26 nm), 0.38–7.43 nm (fluctuation: 7.05 nm), 0.40–7.52 nm (fluctuation: 7.12 nm), and 0.46–7.87 nm (fluctuation: 7.41 nm), respectively.

In addition to inhibiting COVID-19 through its interaction with ACE2, the inhibition mechanism of Mpro is also important for effectively controlling COVID-19. Therefore, stability studies of the binding between the compound candidates and Mpro need to be investigated. The RMSD values of Mpro with licarins A, B, and C and malabaricones B and C seemed to be stable compared to that of Mpro with no ligand. The initial RMSDs after the first 10 ns and the final RMSDs for Mpro-licarin B complex were 1.91 and 1.82 nm, respectively (range: 0.09 nm). The Mpro-licarin C complex had values of 1.52 and 1.70 nm (range: 0.18 nm), and the Mpro-malabaricone C complex had values of 1.44 and 1.58 nm (0.14 nm), respectively. Thus, the RMSD values of licarin B, licarin C, and malabaricone C were smaller than that of the positive control.

3.3. Solvent considerations for extraction of the selected compounds

Eq. (2) was used to find the optimum solubilities of the selected compounds. The highest extraction yields of licarin A, licarin B, malabaricone B, and malabaricone C occurred in acetone, as shown in Fig. 15. DCM extraction yielded the largest amount of licarin C. Although licarin C has lower solubility in acetone than DCM, its solubility in acetone (540 g/L) is nevertheless good. With decreasing iLOGP, solubility showed an increasing trend, as expected. Table 2 shows that the iLOGP values of malabaricone B and malabaricone C indicate that they are the most soluble in acetone, with values ranging from 1,036 to 1,194 g/L, and the iLOGP values of licarins A, B, and C also indicate this, with values ranging from 540 to 792 g/L.

3.4. Solubility of the compounds derived from M. Fragrans

Lipophilicity and hydrophilicity can be used to predict a compound’s solubility because they reflect the single most informative and successful physicochemical characteristic in medicinal chemistry (Malinowska et al., 2019). Therefore, lipophilicity is a critical metric in drug discovery and design (Freeman-Cook et al., 2013). Lipophilicity is expressed experimentally as the partition coefficient (log P), which is the partition equilibrium of the unionized solute in the presence of water and an immiscible organic solvent. Greater lipophilicity is associated with a higher log P value (Arnott
and Planey, 2012). This work used six available models to examine the lipophilicity characteristics of compounds, namely, iLOGP, XLOGP3, WLOGP, MLOGP, SILICOS-IT, and Log Po/w. Four of the six models yielded the same results, which are sorted in descending order by their ability to dissolve in lipids: malabaricone B, malabaricone C, licarin C, licarin B, and licarin A. The solubility results are given in Table 3. The ESOL model showed results consistent with the lipophilicity: licarin A, licarin B, licarin C, malabaricone B, and malabaricone C, ranked from highest to lowest solubility in water.

3.5. Pharmacokinetics of the compounds from M. Fragrans

The pharmacokinetics of the aforementioned drugs are predictably shown in Table 4. Notably, CYP enzymes could be associated with reduced medication clearance from the body. As a result, drug-associated CYP inhibition may play a role in lowering drug metabolism (Almazroo et al., 2017). In Table 4, inhibition of CYP2C19, CYP2C9, and CYP2D6 was shown to be related to increases in the licarin A, licarin B, licarin C, and malabaricone B contents, helping to keep those compounds in the bloodstream
longer by reducing drug metabolism in the liver. Licarin B and malabaricone B can both inhibit CYP3A4, whereas malabaricone B and malabaricone C can also inhibit CYP1A2.

3.6. Drug-likeness of the phytoconstituents of M. Fragrans

Fig. 16 depicts the oral bioavailability predictions for the compounds based on six physicochemical characteristics. The red lines depict the properties of licarin A, licarin B, licarin C, malabaricone B, and malabaricone C, while the pink area represents the ideal range of these features (Lovering et al., 2009, Ritchie et al., 2011). The red lines of three compounds (licarins A, B, and C) in Fig. 16-A-16C lie within the pink area. As a result, we can deduce that these compounds are orally bioavailable.

4. Discussion

The M. fragrans phytochemicals under consideration here must have a lower binding energy value than that of the positive control after analysis by each software utilized. The results showed that a
total of five active substances from *M. fragrans* had lower binding energies than the positive control, as shown in Table 1. Binding pocket analysis of ACE2 assessed by CavityPlus software revealed that ACE2 had a total of 15 pocket binding sites, as shown in Fig. 1A. However, only one of these sites had strong druggability, as shown in Fig. 1B, and the other 14 were found to have low druggability. Three compounds, licarins A, B and C, were able to bind to the highly druggable site of ACE2, as shown in Fig. 1C. In addition to binding to the proper position of ACE2, all three compounds were able to bind to the target protein strongly based on the chemical interactions. Licarin A can form 2 hydrogen bonds (LYS363) and 4 hydrophobic interactions (ALA153, PRO346, and PHE274) with ACE2, as shown in Fig. 3. Licarin B can form 6 hydrophobic interactions (PHE274, ASP269, ALA153, GLU145, LEU144, and CYS344) with ACE2, as shown in Fig. 4. Licarin C can form 5 hydrogen bonds (ASN51, ASN508, TYR510, SER44, and TRP349) and 5 hydrophobic interactions (VAL343, PHE504, LEU120, PHE40, and TRP349) with ACE2, as shown in Fig. 5. In addition to the inhibition of COVID-19 via ACE2, there is a main protease (Mpro) inhibition mechanism that plays an important role in suppressing COVID-19. In this study, binding pocket analysis displayed a total of 9 binding sites on Mpro, as shown in Fig. 6A. However, only one of the Mpro sites had medium druggability, as shown in Fig. 6B, whereas the other 8 had low druggability. Three compounds, licarins A, B and C, were able to bind to the medium druggable site of Mpro, as shown in Fig. 6C. Two compounds, malabaricone B and C, were able to bind to sites with low druggability. The abilities of these compounds to suppress COVID-19 through Mpro binding were mediated by several interactions with the target protein. Licarin A can form 7 hydrogen bonds (CVS145, GLY143, LEU141, SER144, HIS163, and PHE140) and 8 hydrophobic interactions (LEU167, GLN189, PRO168, MET165, HIS41, HIS163, and HIS172) with Mpro, as shown in Fig. 8. Licarin B can form 1 hydrogen bond (THR26) and 6 hydrophobic interactions (THR25, CV5145, HIS41, and MET165) with Mpro, as shown in Fig. 9. Licarin C can form 1 hydrogen bond (GLY143) and 7 hydrophobic interactions (MET165, HIS41, LEU27, CV5145) with Mpro, as shown in Fig. 10. Malabaricone B can form 3 hydrogen bonds (ASP153, ARG298, and ASN151) and 5 hydrophobic interactions (ILE152, PRO9, and PHE294) with Mpro, as shown in Fig. 11. Malabaricone C can form 7 hydrogen bonds (GLN110, GLN127, ASN151, ASP295, PHE294, THR111, and THR292) and 5 hydrophobic interactions (ILE249, PRO293, and PHE294) with Mpro, as shown in Fig. 12. Although the binding sites of malabaricone B and C with Mpro are sites with low druggability, the binding of both compounds with Mpro involves abundant hydrogen bonds. This reflects the ability of both compounds to strongly bind to Mpro. Thus, it was concluded that licarins A, B, and C and malabaricones A and B have the potential to inhibit Mpro.

Although molecular docking studies of the binding between compound candidates and the target proteins may not be sufficient, the use of MD is very important. The findings here demonstrate that these compound candidates bind stably to ACE2, suggesting that they could act as inhibitors. Furthermore, changes in the protein residues upon binding to the compound candidates were identified, indicating that they were not very flexible. To infer good binding affinities, stable RMSD and RMSF values are needed. The residual atom fluctuations of ACE2-panduratin A (positive control) were found to be in the range of 0.56–8.95 nm (fluctuation: 8.39 nm), as illustrated in Fig. 13B. On the other hand, licarins A, B, and C had ranges of 0.49–6.02 nm (fluctuation: 5.53 nm), 0.51–7.85 nm (fluctuation: 7.34 nm), and 0.50–6.36 nm (fluctuation: 5.85 nm), respectively. During the simulation period, the ACE2 fluctuations with licarins A, B, and C were minimal compared to that of the positive control. This result might allow the deduction of their high binding affinity during the simulation period, demonstrating that these compound candidates bind stably to Mpro and suggesting that they could be beneficial as inhibitors. Furthermore, changes in the protein residues were identified upon binding to the compound candidates, indicating that these residues were not very flexible. To infer good binding affinities, stable RMSD and RMSF values are needed. The residual atom fluctuations of Mpro bound to panduratin A were found to lie in the range of 0.47–9.52 nm (fluctuation: 9.05 nm), as illustrated in Fig. 14B. In contrast, licarin A, licarin C, malabaricone B and malabaricone C were found to have ranges of 0.45–7.71 nm (fluctuation: 7.26 nm), 0.38–7.43 nm (fluctuation: 7.05 nm), 0.40–7.52 nm (fluctuation: 7.12 nm), and 0.46–7.87 nm (fluctuation: 7.41 nm), respectively. During the simulation period, the Mpro fluctuations after binding licarin A, licarin C, malabaricone B and malabaricone C were minimal, in the range between those for no ligand and with panduratin A. Thus, these results indicated high binding affinities for the abovementioned compounds during the simulation period.

Our results further demonstrated that acetone was a good choice for *M. fragrans* extraction to obtain large quantities of the five active compounds. These compounds had moderate solubility in water, and all are suitable for oral delivery due to their high GI absorption and CYP inhibition.

To select oral drug candidates, drug-likeness was determined based on the physicochemical properties of the compounds (Daina et al., 2017). Rule-based filters were divided into the following four categories.

1) The following parameters are included in Lipinski's filter: number of hydrogen bond acceptors of less than or equal to 10, number of hydrogen bond donors of less than or equal to 5, MLOGP (lipophilicity) of less than or equal to 5, and molecular weight of less than or equal to 500 (Lipinski et al., 1997).

2) The following parameters are included in Veber’s filter: total polar surface area of less than or equal to 140 and a number of rotatable bonds of less than or equal to 10 are included in Veber’s filter (Veber et al., 2002).

3) The following parameters are included in Egan’s filter: total polar surface area of less than or equal to 131.6 and MLOGP (lipophilicity) of less than or equal to 5.88 are included in Egan’s filter (Egan et al., 2000).

4) The following parameters are included in Muegge’s filter: number of hydrogen bond donors of less than or equal to 5, number of hydrogen bond acceptors of less than or equal to 10, number of rotatable bonds of less than or equal to 15, number of heteroatoms of more than 1, number of carbon atoms of more than 4, number of rings of less than or equal to 7, total polar area of less than or equal to 150, XLOGP3 (lipophilicity) o between −2 and 5, and molecular weight between 200 and 600 (Muegge et al., 2001).

Table 5 shows the results of the drug-likeness evaluation of licarin A, licarin B, licarin C, malabaricone B, and malabaricone C, and we can deduce the following points.

1) All of the compounds (licarin A, licarin B, licarin C, malabaricone B, and malabaricone C) were in accordance with Lipinski’s rule (Chen et al., 2020).

2) All of the compounds were in accordance with Veber’s and Egan’s rules (Oyinloye et al., 2021).

3) All of the compounds, with the exception of malabaricones B and C, which have lipophilicity (XLOGP3) values greater than 5, satisfy Muegge’s criteria (Muegge, 2003).

The general study of the antiviral activities of alternative medicines has been used in the past for treatment. Most of the herbs used to treat viruses have a bitter taste or can be studied from a similar genus. The herb used in this research is one from a similar genus. The herb used in this research is one from a similar genus. The herb used in this research is one from a similar genus. The herb used in this research is one from a similar genus. The herb used in this research is one from a similar genus. The herb used in this research is one from a similar genus. The herb used in this research is one from a similar genus.
stance from *T. crispa* was able to treat this coronavirus better than lopinavir. In addition, the highest number of hydrogen bonds formed with *T. crispa* extract was only 3. However, most substances extracted from *M. fragrans* can form more than three hydrogen bonds.

### 5. Conclusion

This study performed molecular docking of compounds extracted from *M. fragrans*, one of twelve herbs in PSD, to the targets SARS-CoV-2 Mpro and the SARS-CoV-2/ACE2 interaction. Licarins A, B, and C and malabaricones B and C had higher binding affinities (less than −7.0 kcal/mol) than other compounds as determined by each program utilized. Furthermore, the binding affinities of these five active compounds were higher than that of panduratin A, the natural compound and positive control. Both ACE2 and Mpro were predicted to be inhibited by licarin A and licarin B. Mpro was predicted to be inhibited by licarin C, malabaricone B, and malabaricone C. Licarin A and licarin B bind to the same ACE2 location as panduratin A, according to the active site study. Although licarin C binds to ACE2 in a different position, it has a low binding affinity.

As a result of their frequent involvement in ligand–protein binding, ALA153, PRO346, PHE274, and PRO346 are among the most important amino acid residues in ACE2. Licarins A, B, and C bound at the same Mpro site as panduratin A. Malabaricones B and C bound to Mpro at sites different from that of panduratin A, resulting in low binding affinities. At the active binding site, THR26, GLY143, and MET165 are among the important amino acids of Mpro. Furthermore, during the MD simulations period, licarins A, B, and C stably bound to ACE2 with low RMSD and RMSF values. Additionally, licarins A, B, and C and malabaricones B and C showed strong binding affinity to Mpro with low RMSDs and RMSFs during the simulation time.

Most of the compounds (licarin A, licarin B, licarin C, malabaricone B, and malabaricone C) met the criteria of Lipinski, Veber, and Egan, and all but malabaricone B and C met the Muegge recommendations. According to the oral bioavailability predictions, these early data imply that, with the exception of malabaricone C, these compounds might be utilized to treat COVID-19 orally.

Acetone was used to extract *M. fragrans*, which yielded high levels of licarins A, B, and C and malabaricones B and C. As a result, the acetone extract of *M. fragrans* may aid in the prevention of COVID-19. However, more in vivo experimental research is needed to confirm our findings and develop more effective compounds for COVID-19 prevention and treatment.

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### Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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