Hydrogen-diesel fuel co-combustion strategies in light duty and heavy duty CI engines

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ABSTRACT

The co-combustion of diesel fuel with H2 presents a promising route to reduce the adverse effects of diesel engine exhaust pollutants on the environment and human health. This paper presents the results of H2-diesel co-combustion experiments carried out on two different research facilities, a light duty and a heavy duty diesel engine. For both engines, H2 was supplied to the engine intake manifold and aspirated with the intake air. H2 concentrations of up to 20% vol/vol and 8% vol/vol were tested in the light duty and heavy duty engines respectively. Exhaust gas circulation (EGR) was also utilised for some of the tests to control exhaust NOx emissions.

The results showed NOx emissions increase with increasing H2 in the case of the light duty engine, however, in contrast, for the heavy duty engine NOx emissions were stable/reduced slightly with H2, attributable to lower in-cylinder gas temperatures during diffusion-controlled combustion. CO and particulate emissions were observed to reduce as the intake H2 was increased. For the light duty, H2 was observed to auto-ignite intermittently before diesel fuel injection had started, when the intake H2 concentration was 20% vol/vol. A similar effect was observed in the heavy duty engine at just over 8% H2 concentration.

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Introduction

Policy makers and the general public alike are increasingly aware of, and concerned by, the negative impacts of IC engine exhaust gas species on urban air quality and human health [1]. Recent attention has focused on levels of both nitrogen oxides (NOx) and particulate matter (PM) emitted by diesel engine powered vehicles, with higher mortality rates being attributed to PM emissions alone in urban cities. Cities across the world are now moving to establish ultra-low emission zones, and four major cities worldwide have announced plans to ban diesel powered vehicles from 2025 [2]. Future fuel and combustion strategies for IC engines must therefore be designed such that the necessary reductions in pollutant emissions can be met, while also addressing the need to reduce net greenhouse gas emissions [3–5].
Mitigating the adverse impacts of exhaust pollutants from light and heavy duty diesel engines alike is challenging; both in the context of the range of diesel powered vehicle applications and the inherent complications that arise from after-treatment of compression ignition exhaust relative to that from spark ignition combustion. Large vans, trucks and lorries are required to operate both in urban and extra-urban environments and utilise diesel engines of various sizes, the replacement of which with electric and hybrid power units is less imminent than in the case of SI engine powered passenger vehicles [6–9]. Meanwhile, the often locally rich but globally lean stoichiometry of diesel combustion [10] results in elevated levels of PM and excess oxygen in engine exhaust gases that necessitate the use of several after-treatment systems in series for effective reduction of pollutant emissions. One possibility for the reduction of PM formation during combustion [11], and to lower the burden on after-treatment devices, is the use of alternative fuels other than fossil diesel. The use of biofuels is a potential opportunity for the reduction of PM emissions by this approach (in addition to the possibility of reducing GHG emissions from the transport sector) as they often contain oxygen, the presence of which within a fuel molecule can reduce the availability of fuel carbon for soot formation [12], and do not typically contain high sooting tendency aromatic molecules [13]. Fatty acid esters derived from vegetable oils (and commonly referred to as biodiesel) have been widely considered for the displacement of fossil diesel fuel [14]. While the use of biodiesels, either in blends with fossil diesel or unblended, has often resulted in reduced particulate emissions relative to straight fossil diesel, in many studies concurrent increased NOx emissions have also been observed [15,16]. However, studies utilising potential biofuels containing a higher proportion of oxygen than long chain fatty acid esters, for example ethers and short chain alcohols, in blends with fossil diesel and biodiesel have reported simultaneous reductions in both PM and NOx [17,18]. In addition to oxygen bearing fuels, a further approach that has been considered for the reduction of PM in particular has been the partial displacement of fossil diesel or biodiesel with fuels of lower carbon content, for example with co-combustion of natural gas [19] or hydrogen [20–22].

Several H2-diesel fuel co-combustion studies have been conducted in literature with the hydrogen introduced in the intake manifold, and thus aspirated into the combustion chamber with the intake air [23–27]. Co-combustion studies undertaken on naturally aspirated engines have reported reductions in exhaust emissions of NOx and particulates at low H2 substitution levels, however, at higher H2 substitution levels, an increase in both exhaust NOx and particulate emissions has been observed [26–28]. NOx emissions were speculated to increase due to higher in-cylinder gas temperatures resulting from H2 combustion, while particulate emissions increased due to displacement of intake O2 by H2. Some researchers have attempted to mitigate these effects by utilising exhaust gas recirculation (EGR) and intake air boost with H2-diesel co-combustion [29–32]. Miyamoto et al. [29] studied the effect of EGR with H2-diesel co-combustion, and was able to achieve simultaneous reductions in smoke and NOx emissions. Roy et al. [31] managed to achieve a 90% energy substitution of diesel by H2 in a supercharged engine, with N2 utilised as simulated EGR to dilute the intake air. The authors were able to operate the engine at 42% brake thermal efficiency, with negligible levels of NOx and smoke emitted. More recently, Wu et al. [33] investigated the effect of elevated intake air temperature on an engine operating on H2 and diesel fuel. Although a considerable reduction in exhaust emissions was achieved (41% NOx and 30% smoke emissions reduction), no significant impact of intake air temperature on engine performance and emissions was observed by the authors.

More recently, Pana et al. [34] investigated the effect of H2 fuelling on the efficiency and emissions parameters of a truck diesel engine. The authors reported a 10% reduction in brake specific energy consumption (BSEC), a 5.5% reduction in NOx emissions and considerable reduction smoke and soot emissions, with the engine operating on 3.9% energy substitution by H2 relative to diesel only engine operation. This in contrast to the work of Liu et al. [35] who, in aspirating H2 to the intake of a heavy duty diesel engine, observed a reduction in brake thermal efficiency (BTE) and increase in NO2 emissions arising from poor H2 combustion efficiency and significant H2 slip to the exhaust, with the presence of unburnt H2 suggested to have enhanced the conversion of NO to NO2. Lieu et al. [36] also observed low H2 combustion efficiency when aspirated into the intake of a heavy duty diesel engine, and at high loads for the addition of H2 to result in significant increases in peak heat release rates during diffusion controlled combustion. Morgan et al. [37] observed a relatively minor reduction in BTE (~1%) when undertaking H2-diesel co-combustion in a heavy duty engine, attributed to higher diesel fuel direct injection pressures (of up 3000 bar) that enhanced mixing of diesel fuel with the air-H2 mixture and resulted in higher levels of H2 combustion efficiency.

However, H2 combustion in diesel engines can lead to issues such as uncontrolled ignition, surface ignition and backfiring, and high rates of heat release leading to knock. Tsujimura and Suzuki (2017) recently conducted a detailed study on the phenomenon of abnormal combustion (auto-ignition of H2) in a hydrogen-fuelled diesel engine. At low engine loads, the authors did not observe any apparent evidence of H2 combustion in the heat release rate. However, at higher loads and higher H2 fractions, the start of combustion was observed to advance. Abnormal combustion of H2 was observed when the H2 fraction was above 50% under high load engine operation. At these conditions, the cylinder head temperature was also observed by the authors to be strongly dependent on H2 fraction, probably due to increased flame propagation speeds of H2 with increasing H2 fraction.

Hydrogen has also been utilised as a fuel for homogenous charge compression ignition combustion (HCCI), where controlled autoignition of the intake aspirated H2 close to engine TDC is required [38–40]. This was achieved, for example, by Ibrahim and Ramesh [41] in a single cylinder CI engine where H2 was aspirated into preheated air at 120 °C to 130 °C. As the fraction of H2 aspirated into the intake, and thus the equivalence ratio, was increased an advance in the start of combustion (SOC) and higher peak heat release rates were observed, until an engine load of 2.2 bar BMEP (brake mean effective pressure), after which knocking (uncontrolled H2 ignition) occurred.

The above review of literature shows that while a variety of work has been conducted in H2-diesel co-combustion, there is...
still a need for a comprehensive study which looks into the addition of H2 from very low levels up to the levels at which H2 starts autoigniting. It is also important to understand if H2 combustion has similar effects in light and heavy duty diesel engines, or do the different engine geometry, engine load, injection pressures, etc. have a significant effect, and therefore different H2 combustion strategies are required. This paper presents results of H2 diesel fuel co-combustion experiments in both a light duty and heavy duty engine diesel.

**Experimental setup**

The hydrogen-diesel fuel co-combustion tests were carried out at two different experimental facilities, a light duty diesel engine at UCL and a heavy duty diesel engine at University of Brighton. The details of both setups have been described in the following sections.

### Light duty diesel engine

The light duty tests were carried out on a 4-stroke, single cylinder, direct injection compression-ignition engine consisting of a 4 cylinder, 2.0 L Ford Duratorq head, valves, piston and connecting rod from a donor engine, mounted on a single cylinder Ricardo Hydra crankcase. The facility has been described in detail in previous publications [27,30], and Table 1 lists some of the specifications of the setup. The head was fitted with a piezoelectric pressure transducer (Kistler 6056A) coupled to a charge amplifier (Kistler 5018), to measure the in-cylinder gas pressures to a resolution of 0.2 CAD. The in-cylinder pressure was pegged to the intake manifold pressure at the piston intake BDC position every engine cycle. One dimensional thermodynamic models were utilised to derive heat release rates from the measured in-cylinder pressures. Various other pressures (intake manifold, exhaust, etc.) and temperatures (oil, coolant, intake, exhaust, etc.), were also measured and logged onto PCs using National Instruments data acquisition systems (NI DAQ). Diesel fuel was injected into the cylinder using a six-hole, servo-hydraulic fuel injector, with the injection parameters (injection pressure (±1 bar), injection timing (±0.1 CAD) and duration of injection (±1 µs) controlled with an open loop ECU (Emtronix EC-

| Table 1 – Light duty diesel engine specifications. |
|---------------------------------------------------|
| Bore | 86 mm |
| Stroke | 86 mm |
| Swept volume | 499.56 cm³ |
| Compression ratio | 18.3: 1 |
| (geometric) | |
| Maximum in-cylinder pressure | 150 bar |
| Piston design | Central ω – bowl in piston |
| Fuel injection pump | Delphi single-cam radial-piston pump |
| High pressure common rail | Delphi solenoid controlled, 1600 bar max. |
| Diesel fuel injector | Delphi DFI 1.3 6-hole solenoid valve |
| Electronic fuel injection system | 1 µs duration control |
| Crank shaft encoder | 1800 ppr, 0.2 CAD resolution |
| Oil and coolant temperature | 80 ± 2.5 °C |

| Table 2 – Heavy duty diesel engine specifications. |
|---------------------------------------------------|
| Bore | 131.1 mm |
| Stroke | 150 mm |
| Compression ratio | 16:1 |
| Swept volume | 2 l |
| Swirl | Quiescent |
| Combustion chamber | Open Chamber |
| Diesel injection system | Delphi F2E Pumped Injector |

Fig. 1 – Schematic of the light duty diesel engine test facility including the supercharger setup and exhaust analysis systems.
GEN500). H$_2$ was supplied from a compressed gas cylinder supplied by BOC, and fed into the engine intake manifold 350 mm upstream of the intake valves, from where it was aspirated with the intake air into the engine. The flow of H$_2$ was controlled with a Bronkhorst thermal mass flow meter (F-201AV-70K), to an accuracy of ±0.08 l/min over the engine range. N$_2$ gas was supplied from another compressed gas cylinder to the engine intake manifold to simulate EGR-like conditions inside the cylinder by reducing the concentration of intake O$_2$. The flow of N$_2$ was controlled using a separate mass flow controller F-202AV-70K, to an accuracy of ±0.15 l/min over the entire range.

Exhaust gas emissions were collected 300 mm downstream of the exhaust valves and transferred, via a heated line maintained at 190°C, to a Horiba automotive exhaust gas analyser rack (MEXA-9100HEGR) for the measurement of CO and CO$_2$ (non-dispersive infrared absorption analyser), NO$_x$ (chemiluminescence analyser), unburned THC (flame ionization detector) and O$_2$ concentrations (magneto-pneumatic analyser). In addition, the number and size distribution of exhaust gas particulates was measured using a Cambustion particulate spectrometer (DMS500). Fig. 1 shows the schematic of the experimental setup for the light duty diesel engine.

Heavy duty diesel engine

The heavy duty diesel engine tests were conducted on a Ricardo Proteus, described in Ref. [37] and key parameters summarised in Table 2. The combustion chamber was of the quiescent open chamber type, typical of modern heavy duty diesel engines. A Delphi FRE pumped injection system was used for fuel delivery, which was capable of 3000 bar fuel injection pressures. The test facility was modified to convert the engine for hydrogen fumigation of the intake system (Fig. 2). This included a flame arrestor and bursting disk in the intake system of the hydrogen injection system. A Horiba SEC ZSS2MGX mass flow controller was used to regulate the flow of hydrogen into the intake system of the engine. The hydrogen was introduced well upstream of the intake ports in a bend via a perforated tube to promote good mixing with the charge air.

Table 3 – Densities and lower heating values of diesel fuel and hydrogen at 1 atm and 300 K [8].

| Property                   | Diesel fuel | Hydrogen |
|----------------------------|-------------|----------|
| Density (kg/m$^3$)         | 831.9       | 0.0838   |
| Lower heating value (MJ/kg)| 43.14       | 120      |

Fig. 2 – Schematic of heavy duty diesel engine setup.

Fig. 3 – Percentage energy contribution from H$_2$ and the corresponding H$_2$ flow rates (l/min) for the (a) light duty diesel engine tests and (b) heavy duty diesel engine tests.
The charge air was regulated via a set of critical flow nozzles between the air supply reservoir and engine. The engine air mass flow is thus determined by the pressure upstream of the nozzles and is independent of the engine inlet pressure as long as the nozzles remain in the critical condition. This enabled the air mass flow to be held constant as more hydrogen was added to the inlet manifold. Exhaust emission were measured using a Horiba Mexa 7170 and particulate emissions with an AVL 415s smoke meter. Diesel fuel consumption was measured using an AVL 733 fuel balance.

**Experimental methodology**

The same specification of fuel was used across the two experimental facilities. The fossil diesel (RF-06-08 B5) used had 5.3% FAME (fatty acid methyl ester) content, a carbon to hydrogen mass ratio of 6.73:1 and cetane number of 52.7. The compressed H2 used had a purity of 99.995%, while the compressed N2 was of 99.5% purity. Table 3 lists lower heating values and densities of the fuels used in the tests.

**Light duty diesel engine tests**

For the light duty diesel engine tests, the engine speed was kept fixed at 1200 rpm, the diesel fuel was injected at a constant pressure of 600 bar and injection timing of 9 CAD BTDC. The injection timing was selected so that for the diesel only condition, the start of combustion (SOC) occurred at piston TDC.

Two sets of hydrogen-diesel fuel co-combustion were tests carried out on the light duty diesel engine. For both sets, the engine output load was fixed at 6.5 bar IMEP, which was achieved by varying the amount of diesel fuel and hydrogen supplied to the engine for each test to keep the IMEP constant. For the first set of tests, the engine was run in its naturally aspirated mode without any intake air boost or EGR. For the second set of tests, EGR-like conditions were simulated in the engine cylinder by aspirating N2 gas in the intake flow to reduce the O2 concentration in the intake air by 1%. The measured flow rates of N2, H2 and intake air were used to determine the reduction in O2 concentration (v/v) due to the aspirated N2. The temperature of the gas mixture in the intake manifold was 30 ± 1°C for all the tests, measured 200 mm upstream of the inlet valve. Fig. 3(a) shows the percentage ratio of the energy supplied to the engine from H2 to the total energy (i.e., energy from diesel fuel + H2) and the corresponding H2 flow rates.

**Heavy duty diesel engine tests**

For the heavy duty test programme, the A50 key point from the European Stationary Cycle (ESC) was selected. The A50 point on the drive cycle equates to a speed of 1250 rpm and 195 Nm of torque (50% load, 12 bar BMEP) on this engine. Four test conditions were selected; zero EGR and 20% EGR, and then both conditions with the start of combustion phased to TDC and then 10° after TDC. The start of fuel injection was varied to phase the start of combustion to TDC. In all cases, the engine was initially set up to a 25:1 AFR without hydrogen. The air mass flow into the engine was then held constant as hydrogen was added to the combustion chamber. The diesel fuel injection period was reduced to maintain constant load. As more hydrogen is added to the inlet manifold, the inlet manifold pressure would therefore rise but the oxygen available for combustion would remain constant. EGR was also set up with the engine operating only on diesel and measured via the Mexa 7000 by comparison of the CO2 balance between the inlet and exhaust. The back pressure on the engine was maintained at 1.1 bar throughout the programme. The energy contribution from increased quantities of hydrogen for this set up are shown in Fig. 3(b).

**Results and discussion**

**Combustion characteristics**

Fig. 4 shows the in-cylinder gas pressure traces for both the light and heavy duty diesel engine tests. It can be clearly seen from the figure that the peak pressure increase when H2 is added, as a result of H2 co-combusting with the diesel fuel. Fig. 5(a) shows the apparent heat release rates from the H2-diesel co-combustion tests carried out on the light duty diesel engine. The proportion of H2 being added to the intake air was varied (between 0 and 20% H2 v/v), but engine output load (IMEP) was kept constant at 6.5 bar IMEP by adjusting the amount of diesel fuel being injected into the engine cylinder. The results shown in Fig. 5(a) are with N2 also being added to the intake manifold to reduce the intake O2 concentration by...
1% and simulate EGR-like conditions inside the cylinder. It can be seen from the figure that as the proportion of H2 in the intake air is increased up to 20% v/v, the peak heat release rates increase resulting in a greater premixed burn fractions and, therefore, smaller diffusion-controlled burn fractions. The higher peak heat release rates and greater premixed burn fractions could be attributed to the H2 being well mixed with the air, and upon diesel ignition the premixed H2-air mixture (close to the diesel ignition sites) started combusting resulting in high rates of pressure rise and heat release close to the piston TDC position. In comparison, for the diesel only condition, a finite amount of diesel fuel is mixed with the air to combustible limits during the ignition delay period, which then burns resulting in the premixed burn stage. The rate of burning of the remainder of the intake charge depends on the diffusion-controlled mixing of remaining diesel fuel with air. At the highest tested H2 concentration of 20% v/v in air, a reduction in the ignition delay period was observed, which resulted in the peak heat release occurring earlier than the diesel only condition. An interesting phenomenon was observed at the H2 concentration of 20% v/v in air, whereby for some of the engine cycles the pressure rise (and heat release) was seen to occur even before diesel fuel injection had started, as can be seen in Figs. 4(a) and 5(a). Approximately 5% of the cycles underwent this phenomenon. It was thought that this was likely due to the H2 auto-igniting, most probably due to the presence of a hot spot in the engine cylinder.

Fig. 5(b) shows the heat release rate curves for the H2-diesel co-combustion experiments conducted on the heavy duty diesel engine. Above 8% hydrogen (v/v), an aggressive knocking type of combustion was observed, and data was therefore not gathered at these conditions to prevent damage to the engine. Previous studies have statistically investigated the relationship between cyclic variability and various fuel injection parameters [42], and future work could follow a similar approach so as to better understand and potentially control the occurrence of knock with H2-diesel co-combustion. It can be seen from Fig. 5(b) that without H2 addition, the bulk of heat release occurs during the mixing or diffusion controlled burn phase, with a smaller premixed burn phase immediately following the start of combustion. As the H2 concentration is increased, the proportion of energy release during the initial premixed burn phase decreases and the magnitude of peak heat release rate during the second combustion phase (which is diffusion burn dominated at zero H2 addition) increases, and at a H2 addition level of 7.93%, the initial premixed burn phase originally apparent is no longer present. It can also be seen that as the level of H2 addition increases, that peak heat release rates occur earlier during combustion (Fig. 5b), as was also observed in the case of the light duty engine tests (Fig. 5a). In the case of the heavy duty engine tests, it is suggested that the changes in combustion phasing with increased H2 concentration (Fig. 5b) occur due to the increased availability of H2 premixed with air at stoichiometries sufficient for rapid flame propagation following autoignition of the diesel fuel present. It is also possible that the increased rates of heat release rate with increased H2 level may also, in part, be attributable to an increase in the charge density due to increasing boost pressure. It is therefore hypothesised that at the highest H2 addition level of 7.93% in the heavy duty engine (Fig. 5b) that the first phase of

![Fig. 5](image-url)  
**Fig. 5** – Apparent heat release rates for H2-diesel co-combustion in (a) light duty diesel engine and (b) heavy duty diesel engine for different H2 concentrations (by volume) in the intake air.

![Fig. 6](image-url)  
**Fig. 6** – Change in ignition delay (w.r.t. diesel only condition) for H2-diesel co-combustion tests on (a) light duty diesel engine and (b) heavy duty diesel engine for different percentage energy contribution from H2.
combustion, during which the bulk of energy release occurs, is now dominated by heat release from premixed H2, with heat release during diffusion controlled combustion of diesel fuel much reduced relative to the 0% H2 addition case due to the displacement of diesel with H2, and probable increased rates of diesel-air premixing due to heat release from H2 in the initial combustion phase.

It is also possible that the increase in peak heat release rates in the heavy duty diesel engine could be attributable to the observation, that as H2 is added, the bulk of combustion starts to move closer to the piston TDC position. This results in combustion occurring under increasingly lower cylinder volumes and therefore lower energy losses through surface heat transfer. A further possible explanation for the increased rates of heat release apparent is the autoignition of H2 at multiple sites throughout the in-cylinder charge. However, it is interesting to note that if H2 autoignition did occur in the heavy duty engine at 7.93% v/v H2, then this is significantly lower than the level of 20% v/v at which H2 autoignition occurred in the light duty engine. At these conditions, the bulk charge temperature is lower than the autoignition temperature and so other factors must contribute to autoignition. Local hot spots on the combustion chamber wall, or hot particle matter such as soot or oil are inevitable and could act as ignition sites.

The heavy duty engine was of an iron piston and head construction where as the light duty engine had aluminium components. The heavy duty engine would therefore be expected to run at a higher combustion chamber surface temperature due to the iron construction and higher test load, which would explain the earlier transition to autoignition.

Fig. 6 shows the change in ignition delay (w.r.t. diesel only condition) as the amount of energy from the H2 is increased (and the energy contribution from diesel fuel is decreased to maintain a constant engine load output). Ignition delay has been defined as the duration between the start of diesel fuel injection (SOI, the time at which the fuel injector actuation signal is sent) and the start of combustion (SOC), which is determined as the time at which the first detectable incidence of heat release occurs following diesel auto-ignition. For each of the light and heavy duty diesel engine tests, two operating conditions are shown in Fig. 6, the engine being operated without any EGR and then the engine being operated with EGR (in the case of the light duty diesel engine, EGR-like conditions were simulated inside the cylinder by using N2 gas to reduce the intake O2 concentration). It can be observed from Fig. 6 that there is a general trend, in the case of both light and heavy duty engine tests, whereby the increase in H2 energy results in the ignition delay period initially increasing and subsequently decreasing with further H2 addition. The initial increase in ignition delay could be attributed to the aspirated H2 displacing the intake O2, slowing down the low temperature fuel breakdown reactions and delaying diesel fuel auto-ignition. As the H2 concentration in the intake air is increased, the in-cylinder H2-air mixture becomes less lean.

The combustion of this increasingly less lean H2-air mixture, close to engine TDC position results in higher rates of heat release (as can be seen in Fig. 5) and elevated in-cylinder temperatures, leading to a reduction in ignition delay. It could also be speculated that as the in-cylinder H2-air mixture becomes less lean, the H2 flame speed increases, which could...
explain the higher rates of pressure rise and heat release at higher H2 concentrations. The ignition delay results for the light duty engines tests (Fig. 6a) shows that the ignition delay period for the 1% intake O2 reduction condition is generally higher as compared to the only H2 test case. This could also be attributed to the additional N2 further diluting the intake mixture and reducing the O2 available for fuel combustion.

Fig. 7(a) and (b) show the peak heat release rates for the H2-diesel co-combustion tests carried out on the light and heavy duty diesel engines, respectively. It can be seen that in general, the peak heat release rates increase as the H2 energy contribution is increased. The reasons for this – premixed H2-air mixture becoming less lean, leading to higher rates of pressure rise and heat release – have been explained in detail in the discussion for Fig. 5. In Fig. 7(a), peak heat release rates were slightly lower for the 1% intake O2 reduced condition. An exception to this general trend can be observed in Fig. 7(b) for the tests carried out with 20% EGR where the change in H2 concentration does not seem to have a significant effect on the peak heat release rate. This could be speculated to be due to the EGR reducing the intake O2 concentration and lowering the in-cylinder gas temperatures. Therefore, the EGR could most likely offset the elevation of in-cylinder temperatures caused due to H2 combustion, and thereby not net change in the peak heat release rates was observed. It is also worth mentioning the upswing in the rate of heat release curve observed for the no EGR case could still occur at higher hydrogen concentrations which were not studied due to the risk of undesired ignition occurring in the inlet system in the presence of EGR above 4% (v/v).

Fig. 7(c) and (d) show the time of peak heat release rates (tPHRR) for both the test setups. It can be seen from these graphs that the trend of the tPHRR plots follows the trend observed for the ignition delay in Fig. 6. This is expected as the change in the ignition delay period accordingly changes the combustion phasing and hence, the time of peak heat release. Furthermore, similar to the ignition delay trends observed in Fig. 6(a), the time of peak is earlier for the only H2 case, the reasons for which have been explained in the discussion for Fig. 6.

Fig. 8a and b shows the calculated in-cylinder global temperatures with varying H2 addition levels, with and without simulated EGR in the light duty engine. It can be seen from Fig. 8 that an increasing level of intake aspirated H2 results in an increase in peak global temperatures, with and without the presence of simulated EGR, and that the maximum temperature is reached earlier following SOC. This is in agreement with the observed peak heat release rates (Fig. 7a), which increased with the increasing percentage of energy from H2, with and without simulated EGR. Fig. 8b shows very little difference in temperatures reached with and without the use of simulated EGR, at a constant H2 addition levels, which can likely be attributed to the use of N2 aspiration for simulated EGR. Also apparent in Fig. 8 is a trend of increasing temperature from compression only, at approximately TDC and just prior to SOC. This can likely be attributed to the volumetric displacement of intake air with H2, and the subsequent change in the intake mixture ideal gas constant.

Fig. 9 shows the calculated in-cylinder global temperatures with varying H2 addition levels, with and without EGR in the
heavy duty engine. It can be seen that when the engine was operated without EGR (Fig. 9a) that peak in-cylinder global temperatures increased with an increasing level of H₂ addition, as was also observed in the light duty engine tests (Fig. 8a). However, with the use of EGR (Fig. 9b), the effect of H₂ addition on peak temperatures is significantly less pronounced. It is suggested that this can be attributed to the decreased availability of O₂, decreasing rates of additional heat release and temperature increase from H₂, as indicated by the observed peak heat release rates (Fig. 7b). Fig. 9a also shows that as the increase in aspirated H₂ levels resulted in higher in-cylinder temperatures, in-cylinder temperatures subsequently reduced more rapidly than at lower H₂ levels, in agreement with the observed changes in combustion phasing, where the bulk of energy release occurred earlier with increasing H₂ levels. A similar, but much less pronounced, trend can be seen in the light-duty in-cylinder temperatures, where temperatures in the case of the highest H₂ addition level (17.14% v/v) decreased more rapidly to the same level as the diesel only case (Fig. 8a).

Fig. 10 shows the indicated thermal efficiencies with varying proportions of energy from H₂, with and without EGR, of the light duty and heavy duty engines. It can be seen in Fig 10a that there is a decrease in indicated thermal efficiency with an increasing proportion of energy from H₂ in the heavy duty engine. This is in agreement with previous studies [35,36] and can likely be attributed to the comparatively high levels of the aspirated H₂ persisting unburnt to the exhaust gases relative to that of the diesel fuel. However, in the case of the light duty engine (Fig. 10b), such a decrease can only be observed where no simulated EGR (1% O₂ reduction) was included, and an increase in indicated thermal efficiency at levels of energy from H₂ of between 60% and 90% can instead be observed. It is tentatively suggested that this may be attributable to greater detrimental effect of O₂ reduction on diesel combustion efficiency relative to that of H₂ and a concurrent increase in H₂ combustion efficiency due to an increase in the H₂ air equivalence ratio with an increasing proportion of energy from H₂.

Exhaust emissions

Fig. 11 shows the specific exhaust emissions of CO from H₂-diesel fuel co-combustion tests carried out on both light and heavy duty diesel engines. The observed trends in CO exhaust emissions are very similar with both test setups (notwithstanding the range of reproducibility present in Fig. 11b at 0% energy from diesel), in that an almost linear reduction in the gaseous emissions is observed as the percentage energy contribution from H₂ is increased. This reduction in carbon emissions is expected as diesel fuel is replaced by H₂ which does not produce CO as a combustion product.

When comparing the CO emissions between the light and heavy duty diesel engines (Fig. 11), it is interesting to note that CO emissions are lower for the heavy duty diesel engine, as compared to the light duty. This could potentially be due to the higher in-cylinder gas temperatures in the heavy duty engine.
diesel engine resulting in higher rates of diesel fuel oxidation to CO$_2$, and therefore lower CO emissions.

Fig. 12(a) and (b) show the specific NO$_x$ emissions from H$_2$-diesel co-combustion tests carried out on the light and heavy duty diesel engines respectively. First considering the NO$_x$ emissions from the light duty diesel engine (Fig. 12a), it can be seen that as the diesel fuel is substituted by H$_2$, a consistent increase in NO$_x$ emissions is observed. This increase in NO$_x$ emissions agrees with the peak heat release and in-cylinder global temperature trends observed in Figs. 7a and 8a respectively, where an increase in both peak heat release rates and maximum in-cylinder global temperatures was observed with increasing H$_2$ concentration (or increasing percentage H$_2$ energy contribution). This is to be expected as NO$_x$ formation rates in CI combustion are governed by the reactions of the extended Zeldovich mechanism, with conditions of high temperatures and high O$_2$ concentrations resulting in high NO formation rates [43,44]. When considering the NO$_x$ emissions from the heavy duty diesel engine (Fig. 12b), the diesel only levels without EGR are higher than for the light duty case (Fig. 12a), supporting the case that the combustion chamber runs hotter due to the higher load and higher chamber wall temperatures. It might have been expected that the NO$_x$ emissions of the heavy duty with increasing energy from H$_2$ would increase in the same manner (as for the light duty) since the peak heat release rates (Fig. 7b, heavy duty engine tests, only H$_2$ case) and in-cylinder global temperatures (Fig. 9a) both increased with increasing H$_2$ concentration. However, in contrast, NO$_x$ emissions (Fig. 12b) showed a slight reduction with increasing percentage energy from H$_2$. The net NO$_x$ emissions are formed from a large contribution from the diesel diffusion flame and lower contribution from the lean, homogenous combustion of the hydrogen – air mixture. In the heavy duty case, the air-fuel ratio was held constant so which has the effect of making the hydrogen mixture very lean and provides more heat capacity to absorbed the heat released from combustion. The NO$_x$ contribution from homogenous combustion will therefore be low due to high dilution levels and the excess air may reduce the overall temperature rise in the diffusion flame. Notwithstanding the assumptions made in calculating in-cylinder global temperatures, it is suggested that the observed increased rate of temperature decrease following a maximum (Figs. 8a and 9a) provides some support for lower temperatures in the diffusion flame as H$_2$ levels increase. How these complex factors interact is the subject of ongoing research.

Fig. 13a shows the specific exhaust emissions of the unburned total hydrocarbons (THC) in the light duty diesel engine. It can be observed from the figure that THC emissions increase with increasing energy of H$_2$, most likely due to the reduction in intake O$_2$. Fig. 13b shows the specific exhaust emissions of the total particulate mass (TPM) from the H$_2$-diesel fuel co-combustion tests conducted on the light duty diesel engines. It can be seen from Fig. 13b that as the percentage energy contribution from H$_2$ is increased (and that from diesel fuel is decreased to maintain a constant engine load), a considerable reduction in TPM emissions is observed to 25% of their original value when the percentage energy contribution from H$_2$ was increased to 20%. In the case of the
heavy duty diesel engine, at all operating conditions PM emissions were below the instrument detection limit. As discussed in the context of CO emissions (Fig. 11), the reduction in particulate emissions is expected as diesel fuel is replaced by H2.

Conclusions

1. For both light and heavy duty diesel engine tests, the ignition delay was observed to initially increase as H2 was added and subsequently decrease with further H2 addition. The initial increase in ignition delay was attributed to the dilution of the intake air by H2 and N2, while the subsequent reduction was due to elevated in-cylinder gas temperatures from H2 combustion.

2. Peak heat release rates also increased with increasing H2 concentration. This was attributed to the increase in the in-cylinder H2-air stoichiometry leading to high rates of heat release closer to engine TDC position with increasing levels of H2 combustion.

3. In the case of the light duty engine tests, H2 was observed to auto-ignite intermittently before diesel fuel injection had started, when the H2 concentration in air was 20% v/v. It is thought that this phenomenon likely occurred due to the presence of a hot spot inside the cylinder. In the heavy duty engine, significantly increased and advanced rates of heat release were observed at a lower H2 concentration of 8% v/v, possibly aided by the higher operating temperatures of the heavy duty engine relative to those of the light duty engine.

4. NOx emissions were seen to follow the trend of the peak heat release rates and increase with increasing H2 concentration in the case of the light duty engine tests. However, in contrast, for the heavy duty tests, NOx emissions stayed constant/reduced slightly with increasing H2 levels, potentially attributable to lower temperatures during diffusion controlled combustion, despite an increase in peak heat release rates during premixed combustion.

5. CO and particulate mass emissions were observed to decrease as the energy contribution from H2 was increased, and that from diesel fuel (and thus fuel carbon supplied) was decreased to maintain a constant engine load.

It can be seen that while similar effects of H2 intake aspiration, and displacement of energy from diesel fuel, on the duration of ignition delay and combustion phasing were observed in both a light duty and heavy duty engine, the impact of these effects differed. For example, in the light duty engine operating at medium load, NOx emissions correlated closely with peak heat release rates during the premixed burn fraction, whereas in the heavy duty engine at a higher load, NOx emissions showed a greater influence of temperatures during diffusion controlled combustion. Therefore, while H2 has potential for reducing carbon based emissions from both light and heavy duty diesel engines, the degree to which H2 co-combustion can be implemented without undesired H2 auto-ignition or elevated NOx is dependent on engine design and operating conditions.

Acknowledgements

The authors would also like to acknowledge EPSRC (EP/M009424/1 and EP/M007960/1) and Innovate UK (Heatwave II-101560) for their support to this project.

Nomenclature

ATDC after-top-dead-centre
BDC bottom dead centre
BMEP brake mean effective pressure
BTDC before-top-dead-centre
BTE brake thermal efficiency
CAD crank angle degree
CI compression ignition
CO carbon monoxide
CO2 carbon dioxide
EGR exhaust gas recirculation
H2 hydrogen
IMEP indicated mean effective pressures
NOx nitrogen oxides
O2 oxygen
PM particulate mass
ppr pulses per revolution
rpm revolutions per minute
SOC start of combustion
SOI start of injection
TDC top-dead-centre
THC total hydrocarbons

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