Abstract Soil P testing has been widely used to predict crop yields, P uptake, and fertilizer demands in agriculture. Diffusive gradients in thin films (DGT) provides a zero-sink soil P test which mimics diffusion-controlled plant uptake and has previously been found to predict P availability to crops better than conventional quantity-based P tests in highly weathered Australian, though not in European soils. Here we tested the performance of DGT and the Austrian and German standard P quantity test calcium acetate lactate (CAL) to explain the variation of crop yield and P uptake response of winter wheat (Triticum aestivum L.) and spring barley (Hordeum vulgare L.) in long-term P fertilization experiments at four different sites in eastern Austria. Phosphorus extracted with DGT (P-DGT) and CAL (P-CAL) correlated well in similar soils but not across sites with large variation in soil and site properties such as carbonate equivalent and water availability. The predictive power of DGT for barley ($R^2 = 0.42$) and wheat grain yield ($R^2 = 0.32$), and P uptake in wheat grains ($R^2 = 0.36$) was clearly superior to that of the CAL, and less dependent on soil properties. The better performance of DGT compared to the quantity test is consistent with diffusion-limited P uptake in the water-limited cultivated soils of eastern Austria. The critical values of P deficiency derived from the Mitscherlich-type fits for barley and wheat at 80% relative yield are 64.9 and 26.2 μg L$^{-1}$, respectively, consistent with differential P demands of the crops.

Keywords Diffusive gradients in thin films (DGT) · Phosphorus · Calcium acetate lactate extraction (CAL) · Austria · Long-term field experiment

Introduction

Phosphorus (P) is a critical component of cell membranes, nucleic acids, and the energy-rich molecule adenosine triphosphate (ATP), and is central in plant metabolism. Plant P status is therefore critical for maintaining agricultural productivity, which is
reflected by the large P fertilizer inputs to managed soils worldwide. However, P is characterized by strong fixation and slow diffusion in soils, and therefore is often the limiting nutrient for plant growth. Furthermore, due to the unique chemistry of P, the total amount measured in soils bears little relation to the fraction which may be available for plant uptake.

Plant nutrient demands can be expressed either by internal or external requirements. Internal requirements can be defined by the minimum nutrient uptake necessary to obtain a certain yield level or as nutrient concentration in plant tissue that is associated with near maximum yield. External requirements refer to the nutrient availability in the growth medium, i.e., typically soil. The concept of nutrient availability involves four aspects (Fox 1981; Peaslee and Phillips 1981), i.e. (1) capacity, (2) quantity, (3) intensity, and (4) rate of resupply and buffering. Capacity refers to the potential of a given soil to store a nutrient in plant-available (labile) forms, whereas quantity is defined as the realization of this potential, often measured using an extraction method (e.g., Olsen, Mehlich, Bray or CAL) to target this nutrient pool. Intensity has been commonly defined as nutrient concentration in soil solution (Fox 1981). When plants take up nutrients from the soil solution, they deplete the available pool in the rhizosphere. Plant growth/productivity requirements can only be sustained if the rate of resupply, either through desorption from the solid phase and/or diffusion from the bulk soil towards the root surface equals or exceeds nutrient demand. The ability of the soil solid phase to replenish a nutrient has been described by buffer characteristics using the terms buffer power (BP) (Peaslee and Phillips 1981) or distribution coefficient (Kd).

The relative performance of P quantity versus intensity tests has been a matter of debate for some time. Mechanistic modelling and related sensitivity analysis suggest that intensity generally determines uptake in plants more than quantity of P in soil (Barber 1995). However, intensive cropping of ryegrass in pot experiments has also shown that P quantity becomes increasingly important during extended cropping periods (Holford and Mattingly 1976).

According to Nawara et al. (2017), quantity-controlled yield response occurs in sites that have low Kd (defined as the ratio between Olsen P and 0.01 M CaCl2-extractable), while intensity better explains yield response at large Kd values. However, this approach does not consider the extent of saturation of the sorption complex as suggested by Cole and Olsen (1959).

The kinetics of P release from the solid phase have been found to follow one fast and one subsequent slow reaction, assigned to two labile P pools with differential desorption kinetics (De Jager and Claassens 2005; Lookman et al. 1995; Maguire et al. 2001; Menezes-Blackburn 2016; Smolders 2021; Tadesse et al. 2008). A recent experimental and modelling study demonstrates that P desorption rates control P availability to fast growing crops with small specific root area. This is even more relevant in soils with negative P balance as P desorption rates slow down, and may result in decreased P diffusion rates towards plant roots (Smolders et al. 2021).

Apart from differences in the concepts and methodologies to assess external P availability, crop response to soil P is further modified by climate and weather conditions, water availability and P mineralization from organic sources. Moreover, plant roots can actively change their soil environment through various root activities, including release of P-solubilizing compounds. Altogether, these factors and processes create additional variation in crop response that cannot be explained by any soil P test.

Most conventional tests use equilibrium-type extraction to target the quantity (e.g., Olsen or CAL) or intensity (e.g., water or 0.01 M CaCl2 extraction) of soil P for determining external P availability. Diffusive gradients in thin films (DGT) provide a more sophisticated sampling strategy that integrates quantity, intensity and the rate of resupply into one flux measurement in zero-sink conditions, where there is a continuous high concentration gradient between the solid phase and soil solution.

Similar to plant uptake under diffusion limitation, DGT deployment perturbs the solid-solution equilibrium, thus measuring the diffusive supply of an element as well as the resupply from the solid phase. Consequently, strong correlations have been observed between elemental concentrations in plants and those measured by DGT (Zhang et al. 2001, 2004; Song et al. 2004; Koster et al. 2005; Nolan et al. 2005). More recently, DGT has been applied as a tool to predict yield response and assess fertilizer requirements in agricultural systems. Experimental evidence suggests that DGT may be a useful tool to predict P deficiency.
and yield responses to P fertilization in strongly weathered soils of Australia, Africa and Vietnam (Menzies et al. 2005; McBeath et al. 2007; Mason et al. 2010a, b; Six et al. 2013). However, a recent study on temperate European soils indicates that DGT performs at best equal but not better in predicting yields and P uptake in various crops than some conventional soil P tests such as Olsen or ammonium lactate (AL) (Nawara et al. 2017). In field experiments in Denmark and Norway, Mundus et al. (2017) found stronger predictive power of DGT as compared to Olsen, AL and ammonium acetate for P concentration in the youngest fully emerged leaves of barley (*Hordeum vulgare* L.) sampled 30 days after sowing. However, none of the tests, including DGT, were able to predict P uptake and yield at the time of harvest. A pot experiment using soils from Denmark and Southern Sweden found that DGT performed better than Olsen in predicting P uptake in fully emerged young leaves of barley (Tandy et al. 2011).

Overall, there is still limited and confounding information on DGT as a tool for assessing P availability to crops and related critical values below which P deficiency occurs. Here, we follow up on the work discussed above by comparing the performance of DGT and CAL extraction for predicting crop yield and P uptake by addressing the following research questions:

1. In view of its mechanistic advantages, does DGT predict P availability and related wheat and barley yield response better than the Austrian and German standard P test (CAL),
2. does DGT provide an estimate of available P which is less influenced by soil and site conditions, and
3. what are the critical values below which P deficiency occurs.

To answer these questions, we analyzed soil samples of P fertilizer response trials using DGT, located in the main arable farming areas in eastern Austria, and fitted the crop yields and plant P concentrations to the soil test results. We hypothesize that the performance of DGT in predicting grain yields is superior to CAL, and is less affected by soil properties.

**Materials and methods**

**Site and soil characteristics**

Soils were sampled from four different sites (Fig. 1), representing typical soil and climate conditions of the eastern Austrian lowlands. Site and soil description are compiled in Table 1, the ranges of P concentrations extracted by calcium-acetate-lactate (P-CAL) and diffusive gradients in thin films (P-DGT) are shown in Table 2. The western sites in Rottenhaus and Grabenegg are characterized by lower mean annual temperature and higher precipitation as compared to those in Rutzendorf and Fuchsenbigl (eastern sites). This is reflected by the soils, with Chernozems in the latter, and Luvisols / Cambisols in the former sites (Table 1). Additionally, there is a clear differentiation regarding carbonate equivalent and soil pH between Rutzendorf/Fuchsenbigl (moderately to strongly calcareous, pH > 7.5) and the Rottenhaus/Grabenegg sites (slightly to non-calcareous, mean pH < 6.5).

**Field trials**

Experiment 1 is part of a long-term P fertilization study, initiated in 1956 by the Austrian Agency for Health and Food Safety (AGES). The experiment consists of two experimental sites, one in Fuchsenbigl (EXP 1 Fuchsenbigl) and the other in Rottenhaus (EXP 1 Rottenhaus). Utilizing randomized block design, both sites included five replicates of five different P fertilization treatments: no P addition, single superphosphate (SSP) and ‘basic slag’ (Thomas slag) at P application rates of 44 kg ha$^{-1}$ a$^{-1}$ and 175 kg ha$^{-1}$ a$^{-1}$. A detailed description is included in Spiegel et al. (2018). The plots were fertilized continuously from 1956 to 2004, and again in 2012. Both surface soil samples (0–25 cm depth) and winter wheat (*Triticum aestivum* L.) data (i.e., P concentrations in grain and grain dry matter yield) were obtained for the year 2013.

Experiment 2 is a long-term P fertilization study, which investigates different crop residue management strategies. Commenced in 1982, the study is comprised of one experimental site in each of the Rutzendorf (EXP 2 Rutzendorf) and Grabenegg (EXP 2 Grabenegg) locations. Plots were fertilized with SSP at rates of 0, 33, 67, 133 kg P ha$^{-1}$ a$^{-1}$. The fertilization rates were duplicated for both residue
incorporation and residue removal treatments. A detailed description is given in Spiegel et al. (2018). A total of four replicates were produced for each treatment. Yield data for barley (*Hordeum vulgare* L.), and the corresponding surface soil samples (0–25 cm) were collected in 2014.

**Determination of soil properties**

The soil characteristics presented in Tables 1 and 2 were analyzed according to Austrian standard procedures for soil testing. Soils were classified according to IUSS Working Group WRB (2014). Soil texture was determined using a combined sieve and pipette method (ÖNORM L1061-2 2019), the data were used to assess the textural class according to FAO (2006). The pH was determined in 0.01 M CaCl₂ solution with soil to solution ratio of 1:2.5 (v/v) (ÖNORM EN 15933 2012). Soil organic carbon (SOC) was determined through combustion at 650 °C (ÖNORM L 1080, 1999), and the CaCO₃ equivalent was determined according to the Scheibler method (ÖNORM L 1084 2006).

### Table 1 Site, soil and yield characteristics of the experimental sites (mean values and standard deviations)

| Site         | Experiment 1 | Experiment 2 | All data         |
|--------------|--------------|--------------|------------------|
|              | Fuchsenbigl  | Rottenhaus   | Rutzendorf       |
|              | (Eastern)    | (Western)    | (Eastern)        |
|              |              |              | Grabenegg (Western) |
| Mean annual temperature °C | 10.3 | 9.5 | 10.3 | 9.5 |
| Mean annual precipitation L m⁻² | 516 | 696 | 516 | 696 |
| Soil group (WRB) | Calci Chernozem | Stagnic Luvisol | Calci Chernozem | Dystric Cambisol |
| Textural class (FAO) | Loam Silt clay loam | Loam Silt loam |
| Sand g kg⁻¹ | 400 | 30 | 260 | 70 |
| Silt | 420 | 670 | 510 | 770 |
| Clay | 180 | 300 | 230 | 160 |
| CaCO₃ equivalent | 133 ± 13 | 0.36 ± 1.4 | 172 ± 2.85 | 0 |
| SOC | 16.7 ± 0.10 | 19.3 ± 1.37 | 20.7 ± 1.10 | 8.53 ± 0.45 | 16.3 ± 4.90 |
| pH(CaCl₂) | 7.53 ± 0.03 | 6.48 ± 0.30 | 7.63 ± 0.05 | 5.45 ± 0.15 | 6.74 ± 0.94 |
| K-CAL mg kg⁻¹ | 16.7 ± 0.10 | 132 ± 13 | 213 ± 45 | 216 ± 26 | 158 ± 52 |
| P-CAL | 145 ± 91 | 52.8 ± 47.9 | 155 ± 91 | 85.3 ± 52.0 | 110 ± 85 |
| P-CAL:P-DGT L kg⁻¹ | 1091 ± 248 | 431 ± 191 | 1160 ± 946 | 462 ± 228 | 782 ± 620 |
| Wheat yield kg ha⁻¹ | 5720 ± 488 | 6660 ± 672 | 5720 ± 488 | 6660 ± 672 |
| Barley yield | 4390 ± 642 | 6160 ± 671 | 5270 ± 1100 |

### Table 2 Critical values above which P deficiency occurs for P-DGT and P-CAL. Data in parenthesis represent the upper and lower limits of the 95% confidence interval

| Experimental unit | Range of extracted P | Critical limits of P deficiency |
|-------------------|----------------------|---------------------------------|
|                   | P-CAL mg kg⁻¹ | P-DGT µg L⁻¹ | P-CAL mg kg⁻¹ | P-DGT µg L⁻¹ |
| All crops and sites | 9.0–351 | 10.8–788 | 24.6 (8.6–42.9) | 44.0 (27.7–64.3) |
| Wheat (all sites)  | 9.0–347 | 23.0–443 | n.d | 26.2 (12.1–45.1) |
| Barley (all sites) | 19.0–351 | 10.7–788 | 44.9 (48.4–52.1) | 64.9 (44.4–89.8) |
| Eastern sites (all crops) | 9.0–168 | 21.5–788 | 59.2 (29.4–97.2) | 46.6 (25.7–75.1) |
| Western sites (all crops) | 30.0–351 | 10.7–651 | 11.8 (0–28.1) | 37.7 (26.6–50.4) |

*Mean 1981–2010 for the nearest station*
Calcium acetate lactate extractable P (P-CAL)

P-CAL extraction, representing an equilibrium-type quantity P test, was performed according to the Austrian standard procedure (ÖNORM L 1087 2006). Briefly, 2.5 g of air-dried and sieved (< 2 mm) soil was mixed with 50 mL extracting solution, consisting of 0.05 mol L\(^{-1}\) calcium lactate, 0.05 mol L\(^{-1}\) calcium acetate, 0.3 mol L\(^{-1}\) acetic acid, pH 4, according to Schüller (1969). Samples were shaken end-over-end for 2 h, after which they were filtered and measured photometrically using the molybdate blue procedure, as described in Zhang (1998).

Plant digestion

To determine P concentrations, wheat grains (Experiment 1) were dried overnight at 105°C and digested in concentrated H\(_2\)NO\(_3\) and H\(_2\)O\(_2\) (5:1 v/v). Phosphorus was measured using ICP-OES.

P sampled using diffusive gradients in thin films (P-DGT)

DGT samplers consist of a stacked assembly of a diffusive gel and a binding gel layers, both created with the same hydrogel formula, containing 15% by volume polyacrylamide and 0.3% by volume agarose derived cross-linker (Quernmore, Lancaster, U.K., www.dgtresearch.com). Diffusive and binding gels were prepared according to Zhang and Davison (1995), and Santner et al. (2010). Diffusive gels were cast by mixing the gel solution with ammonium persulfate (10%) and TEMED catalyst in proportions according to Zhang and Davison (1995), and further processed as described by Santner et al. (2010). The finished ferrihydrite gels were washed in 1 L type 1 laboratory water (0.055 μS cm\(^{-1}\) produced by a TKA-GenPure purification system, Thermo Electron LED GmbH, Niedereibert, Germany) 2–3 times, and stored refrigerated in 30 mmol L\(^{-1}\) NaNO\(_3\).

The DGT samplers were assembled using plastic housing materials (Quernmore, Lancaster, U.K., www.dgtresearch.com) consisting of a backing cylinder and a cap with a 1.7 cm diameter exposure window. The samplers consisted of a 0.4 mm ferrihydrite gel, polycarbonate membrane (pore size 0.2 μm, thickness 10 μm; Nuclepore, GE Healthcare, Freiburg, GER), 0.8 mm diffusive gel disc, and a 0.13 cellulose nitrate filter disc (pore size 0.45 μm, thickness 130 μm; Supor, Pall GmbH, Dreieich, GER). The polycarbonate membrane negligibly affects the diffusion process and prevents the ferrihydrite gel from sticking to the diffusive gel.

For each experimental soil, a pooled subsample was taken, and maximum water holding capacity (WHC) was determined by visual inspection by wetting soil until a paste formed and the surface was glistening. For DGT sampling, a soil paste was created by adjusting samples to 90% of the WHC, and allowing to equilibrate for 24 h at 20°C. Approximately 3 g of soil paste was then applied on top of the DGT samplers and incubated at 20°C for an additional 24 h. After exposure, samplers were carefully disassembled and gels rinsed with HQ water to remove any remaining soil particles. Ferrihydrite gels were then eluted in 5 mL of 250 mM H\(_2\)SO\(_4\). Samples were shaken on a horizontal shaker overnight to elute.

Phosphorus analysis

Phosphorus concentrations in the eluates were determined using the molybdate blue procedure (Murphy and Riley 1962) on a Hitachi U-2000 UV/VIS spectrophotometer (Hitachi High-Technologies Corporation, Tokyo, Japan). The staining reagent was prepared by mixing 10 mL HQ water, 3 mL of 9 mM ammonium heptamolybdate and 1 mL of 4 mM potassium antimony (III) tartrate hydrate. To produce the color reaction, 1 mL of sample, 0.14 mL of the staining reagent, and 0.06 mL 100 mM ascorbic acid were mixed. After 15 to 20 min, the color reaction had finished and the samples were immediately measured at 881 nm using a spectrophotometer.

Curve fitting and relative yield (RY) calculation

To compare grain yield across sites, we determined the maximum yield for each experimental block separately, by fitting a Mitscherlich-type curve between absolute grain yield and the rate of P applied, using the equation:

\[ y = y_0 + a(1 - e^{-bx}) \]

where \( y_0 \) = yield of control plots (0 P), \( y_0 + a \) = the maximum yield attainable, and \( x \) = the P rate applied.
Using this equation, we determined the maximum yield for each of the experimental blocks \((n = 36)\) at each site, and calculated the relative yields for all treatments according to:

\[
\text{RY} (\%) = \frac{y}{y_0 + a} \times 100
\]

where \(\text{RY}\) refers to the relative yield.

For further analysis we only included P-responsive plots approaching a yield plateau in the highest P treatment, with goodness of Mitcherlich-type fits of \(R^2 \geq 0.80\). Plots were considered as non-responsive if the relation between P application rate and yield was weak \((R^2 < 0.80)\), or close to linear within the range of P applied. This is justified as the lack of a strong relation between P fertilizer input and yield response indicates that P was not the most limiting factor of yield, e.g., because of infection with pathogens, and because RY cannot be calculated using the Mitcherlich equation if the response is regular but linear. The approach follows established procedures of previous work (Burkitt et al., 2016; Mason et al. 2010a, b; Speirs et al., 2013). We then evaluated the relationships between soil test P (P-CAL or P-DGT) and yield by fitting a curve to the data from the remaining 16 experimental blocks, corresponding to 58 soil—RY pairs. To investigate differences in this relationship between crops, we also fitted curves for DGT and CAL separately for wheat and barley. Curve fitting was performed using Solver in Microsoft Excel (Version 15.33). Using inverse calculation from the Mitscherlich-type fits, critical values above which P deficiency occurs were obtained for P-CAL and P-DGT from the intercept of the fitted curve at 80% RY, along with the 95% confidence interval. To this end, we first calculated the confidence intervals for the 80% RY, and again applied inverse calculation to obtain the corresponding lower and upper 95% confidence limits for the P indices.

**Results**

**Soil P test results**

Across all experimental sites, P-CAL varies between 9.0–351 mg P kg\(^{-1}\) (Table 2). According to the Austrian fertilization guidelines (Bundesministerium für Land- und Forstwirtschaft, Umwelt- und Wasserwirtschaft, 2017) this covers the full range of P availability classes from very low (< 26 mg P kg\(^{-1}\)) to very high (> 174 mg P kg\(^{-1}\)). The coefficient of variation of P-CAL across all sites is 77%. Across all sites, P-DGT varies between 10.7–788 µg L\(^{-1}\), with a mean of 206 µg L\(^{-1}\), and a coefficient of variation of 97%, covering potentially P-deficient (Mason et al. 2010a, b; Nawara et al. 2017; Speirs et al. 2013) up to luxury P availability levels (Fig. 1).

Whereas the linear correlation between P-CAL and P-DGT is rather weak \((R^2 = 0.52)\) if all data are included, we observe strong linear relations \((R^2 > 0.83)\) as indicated by the slopes of the relations, there is a clear difference between the soils from the eastern \((P-DGT = 1.73 \text{ P-CAL} – 65.3; R^2 = 0.77)\) and western \((P-DGT = 3.80 \text{ P-CAL} – 50.4; R^2 = 0.86)\). Note that P-CAL in the non-calcareous, fine-textured soils of the western sites does not exceed *170 mg P kg\(^{-1}\) whereas approaches *350 mg P kg\(^{-1}\) in the calcareous sandy soils of the eastern locations (Fig. 2). Conversely, the range of measured P-DGT is not related to the carbonate and sand content of the soils (Fig. 2). The average ratio between P-CAL and P-DGT is more than two times larger at the eastern \((\text{Fuchsenbigl and Rutzendorf})\) compared to the western sites \((\text{Rottenhaus and Grabenegg})\) (Table 2). Even though the P-CAL:P-DGT ratio should not be interpreted as Kd, it is indicative of a lower P intensity and resupply in the sandy, calcareous soils of the eastern sites.

**Grain yields in relation to site and soil properties**

Means and standard deviations of grain yields of wheat and barley are displayed in Table 1. Wheat was
grown in Experiment 1, with on average 16% higher yields at site Rottenhaus compared to Fuchsenbigl. Barley grain yields obtained in Experiment 2 varied more pronouncedly among the sites, with 40% higher yields at Grabenegg compared to Rutzendorf. This yield pattern across sites is consistent with related differences in soil and site conditions (Table 1). The lower yields of both crops at the sites Fuchsenbigl and Rutzendorf are associated with lower water availability as indicated by lower precipitation and higher temperatures (Table 1). As estimated from published climate data (Harlfinger and Knees 1999) for the nearby stations Petzenkirchen (western sites) and Fuchsenbigl (eastern sites), the difference of the climatic water balance amounts to \(200 \text{ L m}^{-2} \text{ a}^{-1}\), with a slightly negative balance at the eastern sites (\(-50 \text{ mm}\)). Moreover, these sites are also less favorable in terms of water holding capacity as indicated by the larger sand contents (Table 1).

Correlation of relative grain yield to soil P tests

Figure 3 shows the correlation of relative grain yield to P-CAL and P-DGT, respectively. Data are presented for crops separately (wheat, barley) and all crops together. In all cases, the Mitcherlich-type fits of grain yield are superior for P-DGT compared to P-CAL. As indicated by \(R^2\), additional 16–19% of the variation in the correlation of soil-test P vs. grain yield can be explained by DGT as compared to P-CAL. The correlation between grain yield and P tests is generally stronger for barley as compared to wheat, with P-DGT explaining 42% of the variation of relative yield for the latter. When data for both crops are pooled, P-DGT explains 34% of the relative yield variation while only 18% are explained by P-CAL (Fig. 3).

Similarly, Fig. 4 displays the correlation of relative wheat grain yield to P-CAL and P-DGT including the data of both crops, but presented separately for the sandy, calcareous eastern, and finely-textured, non-calcareous western sites. The goodness of fits is generally weaker at the western sites, with both soil tests explaining 27–28% of the yield variation. At the western sites, P-DGT \((R^2 = 0.50)\) performs clearly better than P-CAL \((R^2 = 0.33)\), explaining additional 17% of the variation.

Response of plant P uptake to soil P tests

Figure 5 shows the relation between P test results and P concentrations in wheat grains (Panels A and B), Panels C and D the corresponding P uptake. Compared to P-CAL, P-DGT performs clearly superior in explaining additional 18% of the variation in P concentration, and additional 27% of P uptake in wheat grain.

Critical values below which P deficiency occurs

Critical values of P deficiency are compiled in Table 2 for both soil tests along with the related ranges of extracted soil P. The confidence intervals (95%) are presented in parentheses. The critical values are presented only for a RY of 80%, because of the large variation at higher RY levels. The critical values for P-DGT are clearly larger for barley as compared to wheat (Table 2). As indicated by the substantial overlap of the confidence intervals, there is no significant difference between the critical values for P-DGT between the western and eastern sites. On the contrary, the sites are clearly different in terms of the critical values for P-CAL (Table 2).
Discussion

P-DGT is less affected by soil chemistry than equilibrium-type CAL extraction

DGT has gained recognition as an accurate method to measure labile species in aquatic ecosystems, and has been applied to soils and sediments to measure the bioavailability of nutrients and toxicants. More recently, the technique has been utilized by researchers to predict plant response to P, among other elements, and as a tool to assess fertilization requirements in P deficient soils. Our study sought to determine whether DGT has the potential to accurately
predict plant-available P across several sites in eastern Austria, and whether it provides any advantages over the standard P extraction protocol used in Austria and Germany (P-CAL).

Consistent with the clear difference between the sandy, calcareous soils of the eastern, and the finely-textured non-calcareous soils of the western sites (Table 1), the linear relation between P-DGT and P-CAL was only moderate ($R^2 = 0.52$) across all sites and experiments. When the data were divided into soils from eastern (Rutzendorf, Fuchsenbigl) and western sites (Rottenhaus, Grabenegg), the goodness of fits for the linear relations improved to $R^2 = 0.77$ and $R^2 = 0.86$, respectively. We obtained further improvements when each experiment was treated separately, with $R^2$ of the linear relations ranging between 0.83 and 0.99, and regression slopes depending on the experimental site. These results indicate that the methods are correlated if there is little variation in soil properties, but results diverge if soil characteristics are more variable. As shown in Table 1, the sites differ largely in soil texture and carbonate equivalent, with low sand and relatively high clay and silt contents, associated with smaller P-CAL concentrations in the non-calcareous soils of the eastern sites. Site Grabenegg differs from the other locations also in terms of distinctively lower SOC content (Table 1).

We observed a striking difference between the soils from the western and eastern sites regarding the range of P extracted by P-DGT and P-CAL. While P-CAL in the soils at the eastern sites increases over the full range of measured P-DGT concentrations, it does not exceed $\sim 170$ mg P kg$^{-1}$ at the western sites, while P-DGT continues to increase (Fig. 2). As shown by Schüller (1969), the initial pH of the CAL extraction solution of 4.1 increases during extraction up to $\sim 4.8$ in the presence of carbonates, but remains near the initial pH in non-calcareous soils. Lower extraction yields in the non-calcareous soils of the eastern sites could therefore be related to the protonation of surface...
sites due to the lower extraction pH, and subsequent decrease of P desorption from hydrous oxides of Al and Fe (Barrow, 2017). Moreover, in the presence of Ca phosphates, the large Ca concentration in the extraction solution is likely to decrease their solubility (Schüller 1969). In contrast, DGT is operating as a zero sink without considerable modification of soil pH or ion concentration, allowing for P desorption that is much less affected by soil chemistry.

DGT outperforms P-CAL in predicting grain yield and P uptake

Previous work has shown that DGT generally outperforms conventional, equilibrium-type batch extractions in predicting crop yields and P uptake in highly weathered soils (Menzies et al. 2005; McBeath et al. 2007; Mason et al. 2010a, b; Six et al. 2013) while this could not be confirmed for less developed European soils (Nawara et al. 2017). Nawara et al. (2017) suggested that due to strong P fixation in highly weathered (sesquioxide-rich) soils, P availability to crops is mainly controlled by intensity and diffusion, which is better mimicked by DGT. Phosphorus availability in less developed soils with low fixation potential, however, appears to be controlled by quantity, explaining why conventional P tests targeting larger P pools perform equally well or better than intensity tests such as DGT in European soils (Nawara et al. 2017).

Our data obtained from long-term experiments in eastern Austria show consistently stronger predictive power of DGT for crop yield and P uptake compared to the quantity-type CAL extraction, explaining additional 16 to 19% of the overall variation of the barley and wheat grain yield responses (Fig. 3), 18% of the P concentration, and 27% of the P uptake in wheat grains (Fig. 5). While both soil test perform only moderately on the sandy calcareous soils of the western sites, DGT clearly outperforms P-CAL at the eastern sites,
explaining 50%, i.e., 17% more than P-CAL of the crop yield variation (Fig. 4). This finding could be related to the limitation of P-CAL in desorbing plant-available P at the lower extraction pH obtained in the non-calcareous soils, which is indicated by the narrow range of P-CAL (Fig. 2). In turn, this suggests that the zero sink that is induced by DGT without modification of the soil pH allows for desorption of a P pool that can be accessed by plants roots. It has been shown that P desorption rates assessed by DGT are less affected by soil pH and texture (Menezes-Blackburn et al. 2016), and that predictions of yield responses by DGT are less influenced by P buffer power than those by soil pore water and P quantity tests (Degryse et al. 2009).

Note that the experimental sites included in the study of Nawara et al. (2017) are from U.K., western France, Belgium, northern Germany, and Sweden, generally representing areas of considerably higher water availability compared to our sites in Eastern Austria (Dezsi et al. 2018). The water availability is defined as the difference between annual precipitation and potential evapotranspiration (Dezsi et al. 2018). The lower water availability during the growing season at our study sites implies that P re-supply to plants from the solid phase was more constrained, rendering intensity and diffusion control more important than at the study sites of Nawara et al. (2017).

We suggest that the superior performance of DGT in the temperate soils of our study can be explained by the mechanistic similarity between DGT and P uptake in plants (Degryse et al. 2009). While Nawara et al. (2017) view P-DGT merely as intensity factor of P availability because of the rather short deployment time of 24 h, we refer to the more common interpretation as zero sink mimicking diffusion-limited P uptake (Degryse et al. 2009; Menezes-Blackburn et al. 2016). While P availability in temperate regions may indeed be controlled to a larger extent by P quantity in bulk soil material, this does not apply to rhizosphere soil, especially if available water is limited (Gahoonia et al. 1994). The literature provides overwhelming evidence for diffusion-limited P uptake based on rhizosphere modelling and experimental work (Barber 1995; Dunham and Nye 1976; Hinsinger et al. 2011). For fast-growing crops, P diffusion may be also limited by low rates of P desorption from the solid phase (Smolders et al. 2021). Diffusion and/or desorption-limited P uptake is further supported by direct observations using chemical imaging consistently showing depletion of total and available P in the rhizosphere of wheat and numerous other crops even in weakly to moderately weathered soils (Hoefler et al. 2015; Hummel et al. 2021; Santner et al. 2012; Kreuzeder et al. 2018; Wagner et al. 2020). Together, this provides strong evidence that P availability is generally limited by diffusion and re-supply from the solid phase.

While the standard deployment time of 24 h is too short to fully account for the resupply potential of many soils (Nawara et al. 2017; Santner et al. 2015), P-DGT was shown to be correlated with the desorption rate constant of the quickly desorbing P fraction (Santner et al. 2015). Moreover, even in soils with a slow-desorbing P pool, the quickly-desorbing P pool assessed by DGT can represent a relevant fraction of the total available pool (Santner et al. 2015; Smolders et al. 2021), and has been shown to be related to the P status of soils (Menezes-Blackburn et al. 2016). The generally weaker performance of the CAL extraction, especially if applied to soils with variable properties probably also reflects the methodological constraints of equilibrium-type extraction procedures as discussed above.

Critical values of P deficiency for P-DGT are sensible to crop P demands but not affected by soil properties

The critical values for P-DGT derived from the Mitscherlich-type fits for barley and wheat grain yield differ considerably from each other (Table 2), reflecting the different uptake behavior and P requirements of these crops (Spiegel et al. 2001). For wheat, the critical value of P deficiency determined by DGT at 80% RY is 26.2 µg L⁻¹ which compares to 59 µg L⁻¹ reported by McBeath et al. (2007) based on Australian field experiments.

The critical values of deficiency based on P-CAL are considerably larger at the eastern sites (Fuchsenbigl, Rutzendorf) compared to the western locations (Rottenhaus, Grabenegg) (Table 2). This is consistent with the distinct differences of P extractability by CAL (Fig. 2), indicating a strong influence of soil chemistry, in particular carbonate content. Such a difference is not observed for the critical values of P-DGT which indicates—consistent with theory—that this soil test is less influenced by soil properties. The critical values of P deficiency derived in our study for P-CAL
(44.9 mg kg\(^{-1}\) for barley, 24.6 mg kg\(^{-1}\) for all crops, Table 2) compare well with the boundary between the classes “low” and “adequate” (46 mg kg\(^{-1}\)) of the Austrian guidelines for fertilization (Bundesministerium für Land- und Forstwirtschaft, Umwelt und Wasserwirtschaft 2017; Jordan-Meille et al. 2012).

Both crops produced clearly larger absolute yields at the western sites (Table 1). This is related to the dryer and warmer climate at the eastern sites (Table 1), and subsequent lower water availability and reduced diffusional P fluxes and uptake rates. While DGT fluxes were measured at 80% water holding capacity, rates of P diffusion are expected to decrease with soil moisture (Olesen et al. 2001), resulting in lower P and other nutrients supply at a given value of P-DGT. Similarly, water availability could be an important factor contributing to the variation of critical values of P deficiency reported by Nawara et al. (2017) for European sites, and, in addition to differential P fixation properties, differences between European (Nawara et al. 2017) and highly weathered soils of Australia and other regions (Mason et al. 2010a, b; Speirs et al. 2013). It follows that P indices could be improved and become more generally applicable if water availability would be considered.

**Conclusions**

This study contributes to a body of work that evaluates novel methods to assess plant available P in soils under field conditions. Specifically, we evaluated the suitability of DGT as a field test to predict plant available P and consequently make fertilizer recommendations to growers. P-DGT is less influenced by soil properties when compared to the Austrian and German standard P test (P-CAL) and provides a more accurate prediction of crop yield and P uptake. We argue that our findings are consistent with P availability limited by diffusion and re-supply in the weakly to moderately weathered but water-limited soils of our study. We also suggest that standard deployment of DGT for 24 h should be considered not only as intensity method but may properly reflect a relevant part of the plant-available P pool also in soils of temperate regions.

Across different sites, the predictive power of P-DGT might be limited by differential water availability as P diffusion is directly affected by soil moisture. A way forward to account for this could make use of pedotransfer functions such as the one of Olesen et al. (2001) to tabulate impedance factors for different moisture contents based on simple soil textural information. This information could be combined with estimates of the climatic water balance, e.g., aridity indices to obtain correction factors for the effect of water availability on the critical P test values.

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**Code availability** Not applicable.

**Declarations**

**Conflict of interest** The authors declared that they have no conflict of interest.

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