In vitro antifungal, anti-oxidant and HPLC analysis of the extracts of *Physalis philadelphica*
**In vitro** antifungal, anti-oxidant and HPLC analysis of the extracts of *Physalis philadelphica*

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**Abstract**

The different parts of *Physalis philadelphica* were evaluated for the antifungal, anti-oxidant and HPLC assay. The methanol extract of leaf showed inhibitory activity against *Pencillium digitatum* (58% ZI), *Aspergillus flavus* (95% ZI) and *Rhizopus oryzae* (96% ZI) at a concentration of 400 ppm. Strong anti-oxidant activities were found in different parts of the plant. The effects of different extraction solvents on the anti-oxidant activity and antifungal activity of the medicinal extracts were observed. The methanol extract from the leaf, stem, fruit and water fractions from the calyx exhibited strong anti-oxidant activity of 94, 89, 88 and 93% respectively. The ethyl acetate extract was found as a best solvent for the recovery of gallic acid in HPLC assay. No correlation was observed between the gallic acid and anti-oxidant capacity of the extracts.

**Introduction**

Fungal infection of grains is common in agriculture depending countries and causes significant loss of food-stuffs by retarding its nutritive value and make them unfit for human consumption (Devi et al., 2001; Marin et al., 1999). Contaminated cereals with fungi are the principal source of mycotoxin (Desjardins et al., 2000; Galvano et al., 2001). A large portion of the peoples in the developing countries is suffered from the health problems associated with mycotoxin contamination of grain and cereal (Majumder et al., 1997). Due to the toxic effect of synthetic fungicide, it cannot be applied directly to grains (Dukic et al., 2004). Therefore, it is the need of the time to search an alternative approach to store the grains with no toxic effect on the human and animal health.

Extracts produced from the different parts of the plants exhibit antifungal, antibacterial and insecticidal activities under lab trial as reported earlier (Khan et al., 2016; Ullah et al., 2015). Secondary metabolites of plants with pesticidal activity are one of the best alternatives to synthetic chemicals, with least toxic effect and ecofriendly (Varma and Dubey, 1999). This led to researchers to screen out the natural flora for the presence of antifungal bioactive compounds.

The anti-oxidants of the plant source have good impact on the human health and reduce the risk of diseases (Gulçin et al., 2003). The frequency of cancer and cardiovascular diseases can effectively lower with the use of plant-derived anti-oxidants (Arabshahi and Urooj, 2007). Extraction of anti-oxidant depends on the nature of extracting solvent. Thus, preliminary screening is necessary to pinpoint the favorable solvents that finalize the optimum antifungal and radical scavenging activity for the extracts.

Now-a-days HPLC assay is widely used for the evaluation of polyphenol content in herbal extracts, due to its sensitivity, repeatability and accuracy (Granger et al., 2011; Khan and Bakht, 2016). In the present study, the different extracts were screened for the presence of gallic acid (polyphenol) using this assay for the aforesaid reasons.
Cited literatures indicated that there were no attempts to investigate the medicinal importance of different parts of *Physalis philadelphica*, though its different parts are used as folk medicine. In the light of this background, the current study was designed to determine the bioactivities and HPLC of different parts of *P. philadelphica*.

**Materials and Methods**

**Collection and identification of plant materials**

Plant material of *P. philadelphica*, was collected from the District Shangla Khyber Pakhtun Khwa, Pakistan. The plant specimen was identified by Prof. Farukh Hussain, plant taxonomist and deposited in the herbarium of the Department of Botany University of Peshawar, Pakistan.

**Chemicals, media and other accessories**

Different solvents (methanol, *n*-hexane, ethyl acetate, *n*-butanol) used in this study were HPLC, gradient grade (≥99.9%). Potato dextrose agar (70139 Sigma-Aldrich), potato dextrose broth (P6685 Sigma-Aldrich), dimethyl sulfoxide (DMSO) (1211006 Sigma-Aldrich), gallic acid (91215 Sigma-Aldrich), 2,2-diphenyl-1-picrylhydrazyl (D9132 – DPPH) and fluconazole were purchased from the local supplier. The rotary evaporator (Rotavapor R-210/R215; BUCHI Labortechnik AG) was used to evaporate solvents. HPLC spectra were recorded with prominence diode array, PDA, detector (Shimadzu SPD-M20A).

**Preparation of crude extracts of stem, leaf and fruit**

The clean plant material was dried in the shade. The dried material was powdered in an electric grinder. Seven hundred gram of dried powder was stirred in the extraction flask containing 3L of analytical grade methanol. These extraction flasks were kept at room temperature and shaken twice daily for 6 days. Whatman filter paper No. 1 was used to filter the solution. The solid residue was then treated three times with 1,500 mL of fresh methanol. The filtrate was dried in a rotary evaporator to obtain the crude methanol extract. The extract was desiccated in a desiccator at 40ºC (Bakht et al., 2011 a, d).

**Fractionation procedure (Figure 1)**

The crude methanol extract was separated into two parts. One part (10 g) was stored in a glass vial to be tested as crude methanolic extracts for antifungal and anti-oxidant activities while the other part (92 g) was dissolved in 300 mL of water, transferred to a separating funnel to obtain the fractions of the crude extract. An equal volume of analytical grade *n*-hexane was poured into the separation funnel, gently upset it, and then keeps it undisturbed for a moment to separate the two layers. The upper layer of *n*-hexane fraction of the crude extract was collected in the flask and the lower aqueous phase was re-extracted thrice with *n*-hexane. All the quantities of *n*-hexane fraction were combined and dried in a rotary evaporator. The *n*-hexane extract was completely dried in a desiccator at about 40ºC and kept in glass vials until used. The same method of fractionation was also applied to other solvents (ethyl acetate and *n*-butanol). The water phase at the end of the fractionation method was dried via rotary evaporator and desiccator and used as water fraction (Bakht et al., 2011c).

**Antifungal activity and anti-oxidant bioassay**

The antifungal evaluation was performed according to the standard procedure (Khan et al., 2016). Fungal species used in this study were *Candida albicans*, *Trichoderma reesei*, *Acremonium alternatum*, *Penicillium digitatum*, *Rhizopus Oryzae* and *Aspergillus flavus*. These were obtained from the Department of Plant Pathology, The University of Agriculture Peshawar. Fungal strains were cultured on the potato dextrose agar media and standardized on potato dextrose broth. The standardized protocol, elaborated elsewhere, was used for the preparation of both media (Khan et al., 2016).

For the measuring of DPPH radical scavenging activity of the extracts, the standard procedure of anti-oxidant activity was recommended (Mensor et al. 2001). The details methodology for the assay can be found in our previous publication (Khan et al., 2016).

**HPLC assay of fractions**

Each fraction was run on HPLC for the quantification of gallic acid, equipped with water diode detector,
(Shimadzu SPD-M20A), dualistic pump and diamonsil column C18 (diameter 4.6 mm; length 250 mm; particle size 2.5 µM). Standard methodology of HPLC assay was used in this study (Khan and Bakht, 2016). Gallic acid peaks were found in the different extracts with Rt = 9.5 ± 0.11 min. These peaks were also confirmed by observing their UV spectra with standard (gallic acid). The concentration of gallic acid in different fractions was measured by applying the equation:

\[ y = 46599 x + 62968; R^2 = 0.998 \]

y represents the area of peaks and x show the concentration of gallic acid.

**Statistical analysis**

Data of triplicate obtained during experiments were calculated and stated as mean ± standard deviation using Microsoft Excel 2010.

**Results**

**Antifungal activity by well diffusion assay**

The antifungal potential of the tested extracts of leaf against *Rhizopus oryzae* is presented in Figure 2A. Only crude methanol extract possessed inhibitory activity against *R. oryzae* at different concentrations among the tested samples. The methanol extracts measuring 96, 93 and 86% inhibition at 400, 200 and 100 ppm concentration respectively. The different solvents extracted samples of leaves were active against *A. flavus* except for water extract (Figure 2B). Crude methanol extract possessed maximum antifungal potency (95% ZI) shadowed by ethyl acetate (64% ZI) and *n*-butanol fraction (44%) at 400 ppm. These extracts also inhibited the growth of fungus at 200 ppm concentration (methanol extract 47% ZI; ethyl acetate extracted fraction 45% ZI; *n*-butanol 41% ZI). On the other hand, the *n*-hexane extracted sample was active at 200 ppm concentration (34% ZI). The data further stated that ethyl acetate, *n*-butanol and *n*-hexane also possess antifungal potency at 100 ppm concentration, and however, the crude methanol extract did not reduce the growth of the tested fungus at the same concentration. Furthermore, crude methanol extract showed inhibitory activity against *Penicillium digitatum* (58% ZI at 400 ppm concentration, 50.6% ZI at 200 ppm and 43% ZI at 100 ppm). The other extracts of the leaves did not possess inhibitory activity against the same fungus. *Trichoderma*, and *Acremonium* strains were resistant to the leaf extracts. Similarly, the different solvent extracted samples of calyx, fruit and stem did not inhibit the growth of the tested fungus at all concentration.

**DPPH radical scavenging activity of extracts**

The radical scavenging activity of the different extracts from the stem is presented in Figure 3A. Methanol extracted fraction and water extracted fraction showed...
the highest radical scavenging activity at 250 µg/mL. The extracts according to the decreasing order of radical scavenging activity were methanol, water, ethyl acetate, n-butanol and n-hexane at the different concentrations. All these extracts possessed anti-oxidant potency at 5 µg/mL concentration (78.2% by water, 79% by methanol, 72% by ethyl acetate, 65% by n-butanol and 48% by n-hexane fraction). The different extracts of the leaves showed anti-radical properties, to varying degree at different concentrations. The crude methanol extract possessed strong anti-radical activity (94%) shadowed by water extracted fraction (91%) at 250 µg/mL concentration. The other extracts were also active at concentration of 250 µg/mL (i.e. ethyl acetate 81%, n-butanol 76% and n-hexane fraction). However, at minimum concentration level (5 µg/mL), the descending order of antiradical activity of extracts was methanol (82%), water fraction (80%), n-hexane extracted fractions (76%), ethyl acetate (59%) and n-butanol (58.3%).

The tested extracts of calyx showed radical scavenging activity (Figure 3B). Among the tested samples, the water extract exhibited strong anti-oxidant activity (93% at 250 µg/mL followed by methanol extract (92% at 250 µg/mL). Other extracts of calyx in order of decreasing potency were ethyl acetate, n-butanol and hexane extract fraction. Furthermore, all these extracts possessed antiradical property at minimum concentration of 5 µg/mL (i.e. water 85%, ethyl acetate 76%, methanol 71.3%, n-butanol 54.3% and n-hexane 51.2%).

Variation in radical scavenging activity was also found in different extracts of fruit. Water extracted fraction and methanol extract showed strong and same radical scavenging activity (88% at 250 µg/mL) shadowed by ethyl acetate extracted fraction (79% at 250 µg/mL). The descending order of antiradical activity of the other extracts were n-hexane (71%) ethyl acetate extract (81%) and n-butanol (71%) at concentration of 250 µg/mL. Only crude methanol extract (82%) and ethyl acetate fraction (72%) exhibited antiradical property at a concentration of 5 µg/mL.

HPLC assay

In vitro evaluation of plant extracts through HPLC revealed the presence of high gallic acid content (95 ± 1.1 mg GAE/g) in ethyl acetate dry extract of stem among all the tested samples (Figure 4). It is almost thrice of the concentration of gallic acid found in ethyl acetate extracted sample of the fruit (34.0 ± 0.9 mg GAE/g). The minimum concentration of 0.04 mg of gallic acid/g of dry extract was observed in n-butanol fraction of the calyx.

Discussion

In the current study, the inhibitory potential of extracts was evaluated against five fungal strains. Among the tested fungal strains, only three (R. oryzae, A. flavus and P. digitatum) were found to be susceptible to few extracts. Here, the potency of different extracts was compared with pure drug, which is still reasonable. There is a possibility that active ingredient isolated from the extracts might be effective as the standard drug or even more effective. The result of the current study revealed that only leaf part of the plant showed antifungal activity at different concentrations. The crude extract of leaf inhibited the growth of R. oryzae and P. digitatum at both minimum and maximum concentration. Other extracts did not show inhibition of these fungal species. The existence of antifungal potential only in crude leaf extract may be due to the synergistic influence of various bioactive compounds in the same extract. The different extracts of leaf except water fraction were also effective against A. flavus at different concentrations. The growth of the fungus was highly inhibited by crude methanol extract shadowed by ethyl acetate and n-butanol fractions. Similar results have also been stated in previous studies like Johnson et al. (2008), Shariff et al. (2006) and Fridous et al. (1990).

The different extracts of various parts of the plant
(stem, leaf, fruit and calyx) were evaluated for antioxidant activity. Analysis of our results clearly stated that leaf extracts possessed strong radical scavenging activity shadowed by extracts from stem, fruit and calyx. This difference in the antioxidant activity is due to the irregular distribution and localization of active compounds among the different parts of the plant. Besides this, the extracting solvent also affects the antiradical scavenging activity of the extract. In the previous studies the different solvents like methanol, acetone, ethyl acetate and propanol have been used for the recovery of antioxidant compounds (Alothman et al., 2009; Mahattanatawee et al., 2006). The selection of the suitable solvents for the optimum anti radical property of the extract is one of the challenging tasks due to the difference in solubility of antioxidant compounds in various solvents. Thus, preliminary screening is an essential to select the best solvent that finalizes the optimum antiradical activity of extract. The current study also pinpointed the suitable solvent for the optimum radical scavenging activity of the different parts of the plant. In the current study, five different solvents (methanol, n-hexane, ethyl acetate, n-butanol and water) were used for evaluating the antiradical property of different parts of the plant. Different extracts of leaf showed varying degree of radical scavenging activity. The methanol extract of the leaf, stem and fruit possessed strong radical scavenging activity as compared to other extracts, and increased with increasing concentration of extract. From these findings, it can be inferred that methanol could be the best solvent for the recovery of antioxidant compounds. For the calyx, the methanol and water extract showed approximately the same activity. Similar results have been reported in previous studies (Singh and Prakash, 2014; Bushra et al., 2009; Molyneux and Songkla-karin, 2004).

Studies of the gallic acid content through HPLC assay revealed the presence of high concentration of gallic acid in ethyl acetate extracts among the tested samples. The finding of high concentration of gallic acid in ethyl acetate solvent from the different parts of the plant showed good agreement with the results of Janovik et al. (2011), Boligon et al. (2009) and Schubert et al. (2007). Our finding showed no correlation between the gallic acid content and antioxidant capacity of the different parts of the plant. This can be explained by different features, including the presence of various bioactive compounds, their synergistic effect, structure feature of phenolic group and change produced by glycosylation (Jayaprakasha and Patil, 2007; Cho et al., 2003; Yu et al., 2002; Kahkonen et al., 1999).

**Conclusion**

Ethyl acetate was the best choice among the different solvents for the recovery of gallic acid from *P. philadelphica*.

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**Conflict of Interest**

Authors are unanimous in submitting the research paper to journal and there is no conflict of interest.

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