Life Cycle Assessment of Nitrate and Compound Fertilizers Production—A Case Study

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Abstract: The production and utilization of fertilizers are processes with known and noteworthy environmental impacts. Direct greenhouse gas (GHG) emissions and a high contribution to water eutrophication due to the nitrogen (N) and phosphorus (P) derivatives are some of the most crucial impacts derived from the overall life cycle of fertilizer use. The life cycle assessment (LCA) has been a reliable and analytical tool for the identification, quantification, and evaluation of potential environmental impacts of fertilizers related to the products, production processes, or activities throughout their lifecycle. In this paper, a gate-to-gate LCA approach was applied in order to identify and evaluate the impacts derived from the production processes of nitrate and compound fertilizers in the production industry in Northeastern Greece. The results from this study prove that compound fertilizers have a greater impact compared with nitrate fertilizers, contributing up to 70% of the total production impacts. Furthermore, climate change, freshwater eutrophication, and fossil fuel depletion were identified as the most crucial impact categories. Finally, a comparison with relevant LCA studies was conducted, in order to identify the possibility of a consistency pattern of the fertilizer production impacts in general.

Keywords: life cycle assessment; fertilizer industry; nitrate and compound fertilizers; impact assessment; greenhouse gas emissions; cleaner production

1. Introduction

In today’s global economy and consumption needs, chemicals play a vital role as there is no single economic sector where chemicals are not being used. Industries that produce and utilize chemicals have a significant impact on the world’s economy, as the chemical industry is a sector with notable economic growth, providing numerous work positions [1].

The chemical industrial sector includes companies that convert raw materials into different products, which can be classified into the following categories [2]:

(i) The basic chemicals industry, which, besides the production of fertilizers, is responsible for the production of chemicals such as polymers, bulk petrochemicals, and basic industrial inorganic chemicals. Basic chemicals are generally produced in large quantities and are typically inexpensive.

(ii) The life sciences industry, including differentiated chemicals and biological substances, pharmaceuticals, diagnostics, animal health products, vitamins, and crop protection chemicals. These products are usually produced with very high specifications and are closely monitored by government agencies.

(iii) The specialty chemicals (fine chemicals) industry, producing electronic chemicals, industrial gases, adhesives, and sealants, as well as coatings, industrial and institutional cleaning chemicals, and catalysts.

The fertilizer industry (FI), as part of the basic chemicals industry, has a crucial impact on the global economy. In 2018, FI directly employed around 964,000 employees, with East Asia being responsible for approximately two thirds of the global employment status.
Furthermore, the sales revenue of FI in 2018 was estimated to be around $150 billion, which is approximately up to 6% of the total sales of the chemical industry sector. Nitrate fertilizers are responsible for almost half of this sales revenue, followed by phosphate fertilizers (Figure 1) [3].

![Figure 1. Sales revenue of the fertilizer industry in 2018 [3].](image)

Besides the importance for the global economic sector, the production of fertilizers could have a significant impact on the environment, mostly because of the significant amount of energy and natural gas needed for the production processes. According to the data from the European Commission, FI is responsible for the consumption of 2–3% of the total global energy, most of which is used for the production of nitrate fertilizers. This amount of energy and natural gas consumption is used during the intermediate stage of ammonia production [4]. In specific countries, this percentage could be even higher, e.g., in Sweden, chemical fertilizer production and use accounts for up to 20% of the total energy consumed for agricultural purposes [5].

In this paper, the environmental impacts of a FI in Northeastern (NE) Greece are assessed alongside the application of the life cycle assessment (LCA) principles. LCA was developed as an analytical tool, and has become an increasingly common method for the identification and evaluation of potential environmental impacts related to the products, production processes, or activities [6–8]. Impact assessment in LCA traditionally focuses on environmental impacts that derive from emissions, wastes, resource use, and energy consumption, and impacts are categorized by practitioners in global warming potential (GWP), eutrophication, stratospheric ozone depletion, photo oxidant formation, resource use, land use, and others [9]. Moreover, nowadays, LCA is considered one of the best approaches for sustainability assessment in different industrial sectors [10].

LCA is considered an appropriate tool to assess the environmental impacts of the fertilizer production processes. It has been characterized as a suitable method of assessment, as it takes into consideration all of the relevant impacts derived through the entire life cycle of production, in addition to assessing potential environmental improvements [11]. From several LCA studies [12,13], it is evident that fertilizers are responsible for high emissions to the environment, especially those associated with greenhouse gases (GHG), although most available LCA studies do not focus on fertilizer production, instead they consider their use and effects in the agricultural sector. Besides GHG emissions, fertilizers have been linked with impacts such as nitrate leaching into ground water, toxic heavy metal pollution of soils, and aquatic eutrophication due to the use of nitrogen and phosphorus [14].
The selection of efficient raw materials is a means to mitigate the environmental impacts resulting from the production of fertilizers. Decreasing the nitrogen and phosphorus content of fertilizers may result in significantly lower impacts regarding GHG emissions, as well as resource depletion [11]. Furthermore, both nitrogen and phosphorus fertilizers have a low efficiency during their use phase. This fact means that nitrogen and phosphorus derived from fertilizers are not completely absorbed by plants or crops, and thus relatively high amounts of nitrogen and phosphorus are collected in the soil and are lost to the environment [14]. A relevant study has shown that only 30–35% of the total nitrogen was absorbed by plants, while the rest ended up in the environment, causing significant impacts [15].

Thus, taking into consideration all of the above, in this paper, we attempt to assess and quantify the environmental impacts from the fertilizer industry, while focusing solely on fertilizer production and its direct impacts on the environment. The examined fertilizer production industry (FPI) in this study is currently the only remaining industry associated with fertilizer production in Greece. Its operation began in 1965 in the industrial zone of Kavala, NE Greece (Figure 2), and covers an approximate area of 2.600 Ha. Currently, five production units are in operation (Table 1).

![Figure 2. Location of the operating facilities.](image)

Table 1. Production units and final product quantities of fertilizer production industry (FPI).

| Production Unit                              | Final Product(s) | Maximum Annual Production the Last 15 Years (in Ktonnes) |
|----------------------------------------------|------------------|----------------------------------------------------------|
| Ammonia Production Unit                      | Ammonia *        | 147.5                                                    |
| Nitric Acid and Nitrate Fertilizers Production Unit | Nitric Acid * | 224.7                                                    |
|                                              | Nitrate Fertilizers | 253.0                                                  |
|                                              | Solution of Ammonium Nitrate * | 273.7                                                  |
| Sulfuric Acid Production Unit                | Sulfuric Acid *  | 373.0                                                    |
| Phosphoric Acid Production Unit              | Phosphoric Acid * | 65.6                                                     |
| Compound Fertilizers Production Unit         | Compound Fertilizers (NP, PK, and NPK) | 508.0                                                   |

* for internal consumption, as raw material for the fertilizers production.
2. Materials and Methods

In this paper, the gate-to-gate LCA approach was adopted, with the employment of relevant software (Simapro 7.2). The guidelines and specific requirements of the International Organization for Standardization (ISO) were followed, and more specifically, the principles of the ISO 14040 standard. The aim of this study is to identify and assess the cumulative environmental impacts of fertilizer production, while in parallel, to compare the impacts between nitrate and compound fertilizers.

2.1. Scope of the Study and Functional Unit

The overall framework of this paper can be divided into four parts:

1. Quantification of environmental impacts from the specific fertilizer production industry.
2. Comparative impacts from the fertilizer production industry.
3. Comparative impacts from each type of fertilizer.
4. Comparison with relevant LCA studies on fertilizers.

Based on these findings, a number of auxiliary actions promoting sustainable development and greening the industrial facilities will be discussed.

A gate-to-gate approach was used, taking into consideration all of the production processes, from the entrance of the raw materials to the plant, up to the production of the final product. This means that the inputs of the system include energy and water consumption, and raw materials for all stages of production and for every production unit, as well as for the final outputs of the system, which are the final produced fertilizers. Overall, according to the goals of the study, all production processes taking place within the fertilizer plant are included. The boundaries of the study are presented in Figure 3. The transportation of final products to consumers and the use of the fertilizers are not included, as the study focuses merely on the environmental impacts of the production stage.

The functional unit set was the final production of fertilizers for one year (253 kt of nitrate fertilizers and 508 kt of compound fertilizers). Furthermore, based on the final results, a comparison between the impacts per tonne of nitrate and compound fertilizers was conducted, in an attempt to quantify the impacts derived from each type of fertilizer produced within the industrial facilities.
2.2. Inventory Analysis

2.2.1. Data Acquisition

The input data used for this particular study were obtained from the study accompanying the approved environmental terms of the FPI [16]. The approved environmental terms contain all of the primary data describing the production processes within the facilities of the FPI. Background data regarding the extraction of specific first raw materials and transportation to the facilities were obtained from the Ecoinvent database.

2.2.2. Life Cycle Inventory (LCI)

The LCI data for FPI are summarized in Table 2 (for each production process). All data are based on the production of the year, with the maximum production of the last 15 years. The annual production of nitrate fertilizers was approximately 253 kt, while the annual production of compound fertilizers was approximately 508 kt [17].

2.3. Impact Assessment

The data collected to develop the LCI were translated into potential human health and environmental impacts. In this study, the ReCiPe 2008 method was applied. ReCiPe 2008 offers the ability to assess systems and processes by using both midpoint (targeting the environmental mechanisms) and endpoint (targeting the impact) indicators. The overall objective of the ReCiPe method implementation is to translate the list of LCI results into a number of indicator scores, expressing the relative severity on an environmental impact category. These indicators are determined at two levels: 18 midpoint indicators and 3 endpoint indicators, which are presented in Table 3.

Table 2. Life cycle inventory data for nitrate and compound fertilizer production [16].

| Production Process | Raw Materials       | Quantity | Measurement Unit | Database Used |
|--------------------|---------------------|----------|------------------|---------------|
| Ammonia Production | Natural gas         | 1,255,433,415 | Mcal            | Ecoinvent     |
|                    | Water               | 851,280  | Tonnes           | Ecoinvent     |
|                    | Electricity         | 105,120  | MWh              | Ecoinvent     |
| Nitric Acid Production | Ammonia ¹         | 0.36     | Pieces ²        | Ecoinvent     |
|                     | Water completely softened | 70,000  | Tonnes           | Ecoinvent     |
|                     | Water               | 430,026  | Tonnes           | Ecoinvent     |
|                     | Electricity         | 8760     | MWh              | Ecoinvent     |
| Sulfuric Acid Production | Sulfur            | 124,425  | Tonnes           | BUWAL250      |
|                     | Water completely softened | 417,852  | Tonnes           | Ecoinvent     |
|                     | Limestone (lime, hydrated) | 142,270  | Tonnes           | Ecoinvent     |
|                     | Diatomite           | 12,272   | Tonnes           | Ecoinvent     |
|                     | Water               | 172,954  | Tonnes           | Ecoinvent     |
|                     | Electricity         | 20,148   | MWh              | Ecoinvent     |
| Phosphoric Acid Production | Sulfuric acid ¹     | 0.46     | Pieces ²        | Ecoinvent     |
|                     | Phosphate rock      | 203,052  | Tonnes           | Ecoinvent     |
|                     | Water               | 811,335  | Tonnes           | Ecoinvent     |
|                     | Iron sulphate       | 1220     | Tonnes           | Ecoinvent     |
|                     | Electricity         | 14,892   | MWh              | Ecoinvent     |
| Nitrate Fertilizer Production | Nitric acid ¹      | 0.99     | Pieces ²        | Ecoinvent     |
|                     | Ammonia ¹           | 0.38     | Pieces ²        | Ecoinvent     |
|                     | Limestone           | 32,723   | Tonnes           | Ecoinvent     |
|                     | Dolomite            | 3926     | Tonnes           | Ecoinvent     |
|                     | Ammonium sulphate   | 940      | Tonnes           | Ecoinvent     |
|                     | Coating powder      | 222      | Tonnes           | Ecoinvent     |
|                     | Ammonium nitrate    | 218,866  | Tonnes           | Ecoinvent     |
|                     | Magnesite           | 2505     | Tonnes           | Ecoinvent     |
|                     | Electricity         | 7884     | MWh              | Ecoinvent     |
Table 2. Cont.

| Production Process | Raw Materials | Quantity | Measurement Unit | Database Used |
|--------------------|---------------|----------|------------------|---------------|
| Ammonia ¹          | 0.47          | Pieces ² |                  |               |
| Phosphoric acid ¹  | 1             | Pieces ² |                  |               |
| Sulfuric acid ¹    | 0.38          | Pieces ² |                  |               |
| Nitric acid ¹      | 0.01          | Pieces ² |                  |               |
| Ammonium nitrate   | 74,816        | Tonnes   | Ecoinvent        |               |
| Calcium ammonium nitrate | 5490     | Tonnes   | Ecoinvent        |               |
| Ammonium sulphate  | 2050          | Tonnes   | Ecoinvent        |               |
| Potassium sulphate | 68,072        | Tonnes   | Ecoinvent        |               |
| Potassium chloride | 23,974        | Tonnes   | Ecoinvent        |               |
| Potassium nitrate  | 48.5          | Tonnes   | Ecoinvent        |               |
| Sodium phosphate   | 1000          | Tonnes   | Ecoinvent        |               |
| Nitrogen fertilizer| 692           | Tonnes   | Ecoinvent        |               |
| Zinc sulphate      | 427.6         | Tonnes   | Ecoinvent        |               |
| Magnesium sulphate | 793           | Tonnes   | Ecoinvent        |               |
| Boric acid         | 386           | Tonnes   | Ecoinvent        |               |
| Natural gas        | 73,521,950    | Mcal     | Ecoinvent        |               |
| Water              | 191,708       | Tonnes   | Ecoinvent        |               |
| Magnesite          | 392           | Tonnes   | Ecoinvent        |               |
| Coating powder     | 1673          | Tonnes   | Ecoinvent        |               |
| Quicklime          | 528.6         | Tonnes   | Ecoinvent        |               |
| Lime               | 85.6          | Tonnes   | BUWAL250         |               |
| Monoammonium phosphate | 41,248   | Tonnes   | Ecoinvent        |               |
| Electricity        | 26,280        | MWh      | Ecoinvent        |               |

¹ created as a separate process within the LCA software, using materials from the Ecoinvent database. ² Pieces is a functional unit in the LCA software, for processes created within the software itself.

Table 3. Impact categories and category indicators.

| Impact Category Name                  | Abbreviation | Unit                |
|---------------------------------------|--------------|---------------------|
| Midpoint Indicators                   |              |                     |
| 1. Climate change                     | CC           | kg CO₂ eq.          |
| 2. Ozone depletion                    | OD           | kg CFC-11 eq.       |
| 3. Terrestrial acidification          | TA           | kg 1,4-DB eq.       |
| 4. Freshwater eutrophication          | FE           | kg NMVOC            |
| 5. Marine eutrophication              | ME           | kg PM₁₀ eq.         |
| 6. Human toxicity                     | HT           | kg U₂³⁵ eq.         |
| 7. Photochemical oxidant formation    | POF          | kg SO₂ eq.          |
| 8. Particulate matter formation       | PMF          | kg P eq.            |
| 9. Terrestrial ecotoxicity            | TET          | kg N eq.            |
| 10. Freshwater ecotoxicity            | FET          | kg 1,4-DB eq.       |
| 11. Marine ecotoxicity                | MET          | kg 1,4-DB eq.       |
| 12. Ionising radiation                | IR           | kg 1,4-DB eq.       |
| 13. Agricultural land occupation      | ALO          | m² × yr             |
| 14. Urban land occupation             | ULO          | m² × yr             |
| 15. Natural land transformation       | NLT          | m³                  |
| 16. Water depletion                   | WD           | m³                  |
| 17. Mineral resource depletion        | MRD          | kg Fe eq.           |
| 18. Fossil resource depletion         | FD           | kg oil eq.          |

Endpoint Indicators

| 1. Damage to human health              | HH           | Disability-adjusted loss of life years (DALY) |
| 2. Damage to ecosystem diversity       | ED           | loss of species/yr |
| 3. Damage to resource availability     | RA           | Increased cost ($) |


For the interpretation of the results, the ReCiPe method was selected. ReCiPe is a life cycle impact assessment method that transforms the data from the life cycle inventory into tangible results. It was selected because it includes a broad set of impact categories, and it is widely used in similar studies [18]; however, as with every impact assessment method, there are some compromises. ReCiPe, in particular, lacks a quantitative assessment for terrestrial and aquatic eutrophication and acidification. In addition, the normalization process in ReCiPe is performed partially [19], yet it still functions within the acceptable range set by ISO standards.

Concerning the endpoint indicators, ReCiPe is aimed at damage assessment, deriving from the respective environmental impacts. The results are reflected in eleven impact categories, which, through the use of weighting factors, are finally organized in three parent categories of damage, namely [20]:

- Damage to human health, which is expressed as the sum of years lost and the years that someone lives in disability. This sum is expressed by the DALY (disability adjusted life years) indicator from the World Bank and the World Health Organization.
- Damage to ecosystem diversity, which is expressed as losses of species in a specific geographic location, during a specific time period.
- Damage to resource availability, which is expressed as the extra energy needed for the future extraction of minerals and fossil fuels.

3. Results

The results from the implementation of the ReCiPe midpoint method are summarized in Table 4. Examining the impact ratio of nitrate/compound per tonne, it becomes evident that in most of the impact categories, the production of compound fertilizers has higher impacts than the production of nitrate fertilizers. This fact changes in some impact categories. The most significant impact regarding nitrate fertilizers in comparison to compound fertilizers is in fossil resource depletion. This is because nitrate fertilizers use high amounts of energy and natural gas, contributing to fossil fuel depletion [21]. On the other hand, the most significant impact regarding compound fertilizers in comparison with nitrate fertilizers is agricultural and urban land occupation. This is because phosphate and compound fertilizers contain elements that are the cause of environmental problems. Studies have shown that phosphate and compound fertilizers have an increased concentration of some metals (e.g., cadmium), which could potentially be harmful for soil resources [22].

| Impact Category                  | Unit  | Per Tonne of Nitrate Fertilizer | Per Tonne of Compound Fertilizer | Impact Ratio Nitrate/Compound (per Tonne) | Annual Production |
|----------------------------------|-------|---------------------------------|---------------------------------|------------------------------------------|-------------------|
| Climate change                   | kg CO₂ eq. | 751.00                          | 862.17                          | 0.87                                    | 6.28 × 10⁸        |
| Ozone depletion                  | kg CFC-11 eq. | 0.00                           | 0.00                            | -                                       | 1.90 × 10²        |
| Terrestrial acidification        | kg 1,4-DB eq. | 632.42                          | 545.25                          | 1.16                                    | 4.36 × 10⁶        |
| Freshwater eutrophication        | kg NMVOC | 2.09                            | 3.80                            | 0.55                                    | 2.45 × 10⁶        |
| Marine eutrophication            | kg P₂O₅ eq. | 1.06                            | 4.90                            | 0.22                                    | 2.76 × 10⁶        |
| Human toxicity                   | kg U²⁺,S⁵⁻ eq. | 24.31                          | 101.37                          | 0.24                                    | 5.76 × 10⁷        |
| Photochemical oxidant formation  | kg SO₂ eq. | 3.27                            | 19.09                           | 0.17                                    | 1.05 × 10⁷        |
| Particulate matter formation     | kg P eq. | 1.02                            | 0.89                            | 1.15                                    | 7.13 × 10⁵        |
| Terrestrial ecotoxicity          | kg N eq. | 0.66                            | 0.86                            | 0.77                                    | 6.04 × 10⁵        |
| Freshwater ecotoxicity           | kg 1,4-DB eq. | 0.05                           | 0.28                            | 0.18                                    | 1.54 × 10⁵        |
| Marine ecotoxicity               | kg 1,4-DB eq. | 14.31                          | 12.68                           | 1.13                                    | 1.01 × 10⁷        |
| Ionising radiation              | kg 1,4-DB eq. | 14.03                          | 11.93                           | 1.18                                    | 9.61 × 10⁶        |
| Agricultural land occupation     | m² × yr | 1.23                            | 13.80                           | 0.09                                    | 7.32 × 10⁶        |
| Urban land occupation            | m² × yr | 1.12                            | 14.57                           | 0.08                                    | 7.69 × 10⁶        |
| Natural land transformation      | m²     | 0.22                            | 0.16                            | 1.38                                    | 1.34 × 10⁵        |
| Water depletion                  | m³     | 7.51                            | 18.60                           | 0.40                                    | 1.13 × 10⁷        |
| Mineral resource depletion       | kg Fe eq. | 7.87                            | 54.13                           | 0.15                                    | 2.95 × 10⁷        |
| Fossil resource depletion        | kg oil eq. | 616.61                          | 407.46                          | 1.51                                    | 3.62 × 10⁸        |
According to the results presented in Table 4, it can be inferred that the Global Warming Potential (GWP) of the FPI, which is approximately 628 kt CO$_2$eq, is relatively high and comparable, for instance, to the carbon footprint of the nearby Municipality of Kavala, where the total emissions of GHG are 571 kt of CO$_2$eq [23]. The approximate amount of 628 kt of CO$_2$eq emitted to the atmosphere during the production stages of nitrate and compound fertilizers is mostly CO$_2$, because of the extensive use of fossil fuels in the ammonia production unit and N$_2$O, which is usually produced during the production of nitric acid, serving as a first raw material for the production of ammonium nitrate [11].

In order to facilitate the comparison and interpretation of the results, a normalization step was applied to our results, as shown in Table 4. The normalization step in the ReCiPe method is based on the report of Sleeswijk et al. [24], and recalculates the results per citizen as a common reference impact. This means that the scores for each impact category are all compared to the annual impact of an average European person, which is 1 kpts. In our case, for example, the impact of the total production to freshwater eutrophication is 2.800 Kpts, which is equivalent to the impact of 2.800 average European people. The normalized results are presented in Figure 4. The greater impacts are related to freshwater and marine environments. Freshwater eutrophication and marine ecotoxicity are by far the categories with the highest impacts, followed by freshwater ecotoxicity and natural land transformation.

![Figure 4. Results using the ReCiPe midpoint method after normalization (for the total production, expressed in Kpts, where 1 Kpt is the impact of an average European person).](image)

These high impacts on freshwater and marine environments can be explained through the composition of the specific fertilizers produced. Nitrogen and phosphorus are the core materials for nitrate and compound fertilizers. These two nutrients, however, play a crucial role in the eutrophication of water bodies as well. The enrichment of water environments with nutrients, especially nitrogen and phosphorus, may lead to an undesirable disturbance to the balance of organisms present in the water and to the quality of the water [25].

The contribution of each process to the final fertilizer production is presented in Figure 5, using the ReCiPe endpoint method. Compound fertilizers are responsible for up to approximately 70% of the total impacts, resulting from the amount of processes and first raw material needed for their production.
Figure 5. Schematic contribution of processes using the ReCiPe endpoint method (left: per tonne of compound fertilizer, right: per tonne of nitrate fertilizer).

The ammonia production unit’s contribution to the final impact is among the highest for the production of both types of fertilizers. One of the reasons for this specific result is the high amount of natural gas and energy consumption. As seen in Table 2, the ammonia production unit consumes large quantities of electricity and requires significant quantities of natural gas as a raw material for the production of anhydrous ammonia. This could potentially raise questions regarding the economic and environmental sustainability of the ammonia production unit, which is a worldwide phenomenon. Globally, the production process of ammonia is responsible for roughly 1% of the global energy consumption (approximately $5.6 \times 10^{20}$ J/year) [26], which has led to the need for finding new alternative ways of producing ammonia (e.g., processes at atmospheric pressure and relatively low temperatures, as described by McEnaney et al. 2017) [27].

The results from the endpoint analysis are summarized in Table 5. Damage to human health and ecosystems caused by fertilizer production is mostly attributed to climate change, while fossil depletion is responsible for the damage to resources category.

In order for these results to be easier to understand, the ReCiPe endpoint single score method was used. In it, score, characterization, damage assessment, normalization, and weighting are combined. Although there have been a lot of arguments against using a single score, as it is not always suitable for the life cycle inventory assessment (LCIA; such as the lack of transparency in the value choices that are deployed), the use of single score analysis, combined with midpoint and endpoint indicators, can assist in a better interpretation of the results [28]. In Figure 6, the single score results for both the total production and per tonne of produced fertilizer are presented. The compound fertilizers are responsible for most of the impacts, in both the total production and per tonne of produced fertilizer, proving a consistency in results using the midpoint and endpoint methods. In total, damage from fertilizer production is more intense in human health, mostly because of the climate change and human toxicity impact categories, and in the resources category because of the fossil depletion (results that are consistent with endpoint damage assessment).
Table 5. Results using the ReCiPe endpoint analysis (maximum annual production).

| Damage Category | Impact Category                     | Unit   | Relative Contribution (%) | Total     |
|-----------------|-------------------------------------|--------|---------------------------|-----------|
| Human Health    | Climate Change Human Health         | DALY   | 100.00%                   | 1.90 × 10³|
|                 | Ozone Depletion                     | DALY   | 46.3%                     |           |
|                 | Human Toxicity                      | DALY   | 0.03%                     |           |
|                 | Photochemical oxidant formation     | DALY   | 16.1%                     |           |
|                 | Particulate matter formation        | DALY   | 0.01%                     |           |
|                 | Ionising Radiation                  | DALY   | 37.7%                     |           |
|                 |                                    |        | 0.05%                     |           |
| Ecosystems      | Climate Change ecosystems            | Species.yr | 100.00%     | 7.42 × 10⁶|
|                 | Terrestrial acidification            | Species.yr | 67.11%      |           |
|                 | Freshwater eutrophication            | Species.yr | 0.81%       |           |
|                 | Terrestrial ecotoxicity              | Species.yr | 0.42%       |           |
|                 | Freshwater ecotoxicity               | Species.yr | 0.26%       |           |
|                 | Marine ecotoxicity                   | Species.yr | 0.04%       |           |
|                 | Agricultural land occupation         | Species.yr | 1.16%       |           |
|                 | Urban land occupation                | Species.yr | 2.00%       |           |
|                 | Natural land transformation           | Species.yr | 28.16%      |           |
| Resources       | $                                   | 100%   | 5.83 × 10⁹               |           |
|                 | Metal depletion                      | $      | 0.04%                    |           |
|                 | Fossil Depletion                     | $      | 99.96%                   |           |

![Figure 6. Single score results using the ReCiPe endpoint method.](image)

4. Discussion

Comparing this LCA study to other LCA studies regarding the production of different fertilizer types, we can verify that the results seem to be consistent. Hasler et al. [11], in a relevant LCA study, selected five key impact categories to investigate as the most associated in fertilizer production. Among these categories were climate change, freshwater eutrophication, and fossil fuel depletion, categories that have showed greater impacts in this study. The impact category for climate change, associated with GHG emissions, was proven to have significant importance, mostly because of the CO₂ emissions (mainly from
fossil fuels used in ammonia production and in the reaction of phosphorite with sulfuric acid [6]) and N$_2$O emissions, derived during the production of nitric acid, which is used in nitrate fertilizers and ammonium nitrate production [11,29].

Specifically regarding GHG emissions per tonne of produced fertilizers, there are not a lot of case studies available. Most of the relevant literature focuses on the use of fertilizers and the overall impacts of the agricultural sector. Some available results are compared with our own case study in Table 6.

Table 6. Comparison with results from relevant case studies.

| Case Study            | Type of Fertilizer | GHG Emissions (in kg CO$_2$eq/t Fertilizer) |
|-----------------------|--------------------|---------------------------------------------|
| Dziuba et al. 2016 [30] | Nitrate            | 600–1500                                    |
| Hasler et al. 2015 [11] | Compound          | 1440                                        |
| Kobayashi & Sago, 2001 [31] | Nitrate and Phosphate | 732                                        |
| Brentrup et al., 2008 [32] | Potash            | 260                                         |
| This study            | Nitrate            | 751                                         |
| This study            | Compound           | 862                                         |

The results from the available relevant case studies further validate the results from this study, as for the case of both nitrate and compound fertilizers, our results are within the range discussed in the literature. A significant observation is the notable lower result for potash fertilizers; although these are the results of one study, hence further examination is needed in order to assess the overall impacts compared with different fertilizer types.

The high energy intensity of nitrate fertilizer production is considered to be responsible for the high impact on fossil fuel depletion. In the production of nitrate fertilizers, a significant amount of oil equivalent is needed, mostly from natural gas, during ammonia production. Furthermore, the process where nitrogen is transformed into ammonia is proven to require more than 1% of all of the energy produced worldwide, contributing significantly both to fossil fuel depletion and GHG emissions [33,34]. An idea for resolving the dependency of fertilizer production on fossil fuels is the use of biogas; however, this solution also comes with the burden of increasing the impact on acidification and eutrophication, as presented by Ahlgren et al. in 2010 [35].

5. Conclusions

Fertilizers production is known to have significant environmental impacts throughout the whole cycle of the production stages. The purpose of this LCA study was to determine the total environmental impacts derived from fertilizer production, and to compare the impacts from nitrate and compound fertilizers separately. A case study of an operating fertilizer industry in Greece was selected. This study showcased that compound fertilizer production has higher environmental impacts compared with nitrate fertilizer production, because compound fertilizer production utilizes a wider number of raw materials, as well as materials that are produced through other production processes and/or units within the fertilizer industry. Regarding specific production stages, the ammonia production unit that, in our case, operates within the facilities, supplying ammonia as a raw material both for nitrate and compound fertilizer production, is proven to have significant contributions to the total environmental impact profile, because of its high energy intensity and the relatively high amounts of natural gas required.

As far as the impacts derived from production, the global warming potential is relatively high, with emissions of approximately 630 kt of CO$_2$eq. This high number of GHG emissions, mostly CO$_2$ and N$_2$O, can be explained by the use of fossil fuels in ammonia production and during the production of nitric acid, used as the first raw material for nitrate fertilizers and ammonium nitrate. Furthermore, as expected, because of the use of nitrogen and phosphorus derivatives, freshwater eutrophication and ecotoxicity are issues that should be taken into consideration for the local water environment. Finally,
using the endpoint analysis, we have seen a significant impact in resources depletion, mostly fossil fuel depletion, resulting from the high energy intensity of the production stages (especially in the ammonia production unit).

In conclusion, the goal of this study was to develop an accurate LCA model, describing nitrate and compound fertilizer production as a whole, and identifying and evaluating the potential environmental impacts. The results presented can serve as a starting point for the general consideration of optimizing the production stages of fertilizer production in a more sufficient and sustainable matter. It is proposed that future research broadens the boundaries of the LCA model in a cradle-to-grave approach, in order to assess the overall environmental performance of the fertilizers (both production and use). Under the specific framework, issues such as the consumption and performance efficiency of different types of fertilizers should be considered. As the production of fertilizers has been proven to have high demands on energy and raw materials, as well as emitting significant amounts of GHG, the incorporation of symbiotic activities and the establishment of an industrial symbiosis network in the area of our case study is something worth examining in order to minimize resource consumption and emissions.

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