Keratinase Activity of A Newly Keratinolytic Bacteria, 
*Azotobacter chroococcum* B4

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Abstract

Keratinase is a proteolytic enzyme capable on degrading the hardy polymeric biomolecule or keratin. In recent days, the utilization of keratinolytic microorganisms is seen as a promising way in recycling the keratin wastes from the avian and mammalian into valuable derived products. Previous study has reported the presence of a keratinolytic bacterium, later identified as *Azotobacter chroococcum* B4 obtained from dump soils. The present study investigates the enzyme characteristics of keratinase produced by this strain based on the physical appearance of final degraded product using SEM, the molecular weight of keratinase using SDS-PAGE, the effects of nutrition (C/N-source) on strain production of keratinase and the enzyme stability in metal ions solution. The molecular weight of keratinase produced by *A. chroococcum* B4 was about 30 kDa. Both sucrose and tryptone supplementation increase the keratinase activity by 71.7 and 97.8 U/mL after 96 h of cultivation. Metal ions, Ca²⁺, Mg²⁺, Mn²⁺, Na⁺, and K⁺ are regarded as activators by increasing the relative activity of keratinase by 117, 166, 111, 113, and 112% respectively, while phenylmethylsulfonyl fluoride (PMSF) is regarded as inhibitor by decreasing the relative activity down to 31%. Based on the metal ion characteristics, this strain produced a serine-protease type of keratinase which may further studied for its application in the field.

Keywords: *Azotobacter chroococcum*, keratinase, sucrose, tryptone

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INTRODUCTION

Poultry-processing industries generate more than 4 billion pounds of feathers annually as waste. In chicken feather weight accounts up to 5-7% of its body chicken. Chicken feather contains about 90% of keratin. 

Keratin is a stable and insoluble structural protein tightly packed by the numerous polypeptide chained with disulfide cross-linkage (S=S bonds) which increased the hydrophobic nature of this polymer. It is highly resistant to degradation by conventional proteases. The stability of keratin depends on number of helical chains (∝-keratin) and β-sheet (β-keratin) of proteins and their cystine bridge. Biodegradation of keratin involving microbial ketainase is an easy and economical method for conversion of keratin waste into useful products as source of amino acids and proteins, which are typically utilized as animal feed additives as well as nitrogen source for plants. Keratinase can be utilized as biocatalysts in leather and textile industry, and waste recycling. 

Keratinase is produced by various types of fungi and bacteria, including Gram-positive bacteria of Bacillus and Gram-negative, such as Vibrio sp. strain kr2, Aeromonas hydrophilia strain FB3, Stenotrophomonas spp., Chryseobacterium spp. A fungal isolate from crocodilian feces was reported to degrade chicken feather waste within 10 days. During a screening of keratinolytic bacteria from various sources, we reported a new strain of Azotobacter chroococcus B4 which produced a considerably potential keratinolytic activity. The purpose of this study was then to obtain the biological characteristics of keratinase produced by A. chroococcus strain B4 for a better understanding of its nature in the future application.

MATERIALS AND METHODS

Isolate and growth medium

Azotobacter chroococcus B4 was grown on feather meal broth (FMB) composed of chicken feather powder (15 g), K2HPO4 (0.7 g), KH2PO4 (0.4 g), NaCl (0.5 g), MgSO4 (0.1 g) in 1000 mL distilled water. Growth medium was sterilized at 121°C, 1 atm for 15 min. Chicken feather waste was collected from slaughterhouses. Feathers were washed using detergent-water and cut into small pieces (±2 cm) prior soaking in acetone for 24 h. Washed feathers were dried in oven at 40°C for 72 h and subjected to be grinded to make chicken feather powder.

Observation using scanning electron microscope (SEM)

Whole chicken feathers were inserted into A. chroococcus B4 liquid culture and incubated for 48–96 h, 37°C under agitation of 180 rpm. After incubation, a feather or strand sample was soaked into a 2% (w/v) sodium coccolylate buffer for 1 h. The sample was further soaked into a 1% (w/v) tetraoxide solution for 1 h. The sample was removed and dipped into following EtOH concentrations of: 70, 80, 90 and 100%, each for 30 sec. The sample was removed and immersed into a n-butanol solution prior coating into gold-coated metal plate. The plate was placed inside SEM apparatus operating at 20 kV while tubes were conditioned in vacuum (0 Pa).

Effect of C- and N-sources

Variation in carbon and nitrogen sources was used to promote bacterial keratinolytic activity. Glucose, sucrose, starch, fructose, lactose, and sorbitol (1% w/v) was tested as C-source, while casein, gelatin, KNO3, peptone, tryptone, NaNO3 and yeast extract (0.5% w/v) was used as N-source. pH and temperature were adjusted as described in previous study. Keratinase activity was measured at 24, 48, 72, 96, 120, 144, and 168 hours.

Purification of keratinase

Crude keratinase was harvested from an overnight culture by centrifugation at 8000×g, 4°C for 20 min. Supernatant was precipitated using 70% (w/v) (NH4)2SO4 at 5°C for 24 h and subjected to further centrifugation at 8000×g, 4°C for 20 min. Ten milligrams of pellets were dissolved by adding 5 mL buffer A (2:1, 25 mM Tris-HCl, pH 8) and dialyzed inside a cellulose membrane (10 kDa cut-off). Dialysis membrane was then submerged into 600 mL buffer B (50 mM Tris-HCl, pH 8) following slow stir for 24 h at 5°C. The buffer was periodically changed in the interval of 8 h. Enzyme solution was further purified using a Sephadex G-50 (Sigma-Aldrich) gel filtration chromatography column (fractionation range 1.5 to 30 kDa). Partially-purified protein solutions were gently poured into the column and eluted with 25 mM Tris-HCl, pH 8.0. Fractions were stored in vials and subjected to further experiments.
Keratinase assay

Keratinase assay was performed as previously described\textsuperscript{10} with some modifications. Chicken feather powder was used as a keratin source and substrate in the keratinase assay. The keratin substrate (4 mg) was dissolved in 1 mL 50 mM Tris–HCl, pH 8.0. Keratin solution (250 μL) was reacted with 500 μL enzyme solution in 250 μL 50 mM Tris–HCl buffer. Solution was placed on a hot plate for 30 min at 40°C. The reaction was stopped by mixing 1 mL 10% (w/v) trichloroacetic acid solution followed by incubation for 10 min at 4°C. The reaction mixture was centrifuged for 10 min at 10,000×g. Free polypeptides in the supernatant were measured at λ 280 nm. A standard was prepared using tyrosine. One unit of enzyme activity was defined as the amount of keratinase that caused an increase of 0.01 in absorbance at λ 280 nm.\textsuperscript{12} Measurement of protein content was based on colorimetry method using Bradford reagent. After mixing 500 μL samples and 750 μL Bradford reagent, protein concentration was determined using a spectrophotometer at A\textsubscript{595nm}.

Effect of metal and inhibitor to keratinolytic activity

Effect of metal ions on keratinolytic activity was examined using 5 and 10 mM chloride salts of Ca\textsuperscript{2+} (CaCl\textsubscript{2}), Mg\textsuperscript{2+} (MgCl\textsubscript{2}), Ba\textsuperscript{2+} (BaCl\textsubscript{2}), Na\textsuperscript{+} (NaCl), K\textsuperscript{+} (KCl), Co\textsuperscript{2+} (CoCl\textsubscript{2}), and Mn\textsuperscript{2+} (MnCl\textsubscript{2}). Solutions of keratinase and ions were incubated at 37°C for 1 h in 50 mM phosphate buffer (pH 7.5)\textsuperscript{15}. Inhibitors i.e. pepstatin A, benzamidine, phenylmethylsulfonyl fluoride (PMSF), ethylenediaminetetraacetic acid (EDTA), dithiothreitol (DTT), 2-mercaptoethanol, soybean trypsin inhibitor, N-tosyl-L-lysine chloromethyl ketone (TLCK), bromoacetic acid, chymostatin, and iodoacetic acid were tested. Inhibitor was mixed with purified keratinase at 1 and 5 mM concentration. All reactions were stopped using 500 μL of 10% (w/v) TCA\textsuperscript{16}. Pellet was separated by spinning at 8000xg for 30 min at 4°C. Reaction with no metal and inhibitor addition was used as a control.

Determination of keratinase molecular weight

Molecular weight was measured using SDS-PAGE. SDS-PAGE was conducted in 10% (w/v) polyacrylamide resolving and 5% (w/v) stacking gel. Sample of 15 μL (protein conc. 2 μg/mL) dissolved in a loading dye were incubated at ±100°C for 5 min for protein denaturation. The samples were inserted into wells and run at 110 V for 90 min. After separation, gel was removed and stained using Coomassie® Brilliant Blue (CBB) R-250 in orbital shaker. Consecutively, the gel was soaked in methanol acetate solution (250 mL distilled water, 200 mL methanol, and 50 mL acetic acid). The molecular weight (kDa) of purified keratinase was determined by measuring the relative migration distance (Rf) between protein standards and the unknown protein analysed using linear regression method. The presence of multiple bands was confirmed through Zymogram analysis using keratin as specific substrate for enzyme detection. Zymogram was performed using a 10% separating gel added with 0.2% (w/v) keratin.
powder. The gel was immersed in 2.5% Triton® X-100 for 1 h at 37°C and subsequently in a 50 mM Tris-HCl buffer (pH 8) overnight. The gel was stained using 0.05% (w/v) CBB for 2 h followed by destaining until white band was formed visually.

RESULTS AND DISCUSSION

Chicken feather degradation by *Azotobacter chroococcum* B4

Newly keratinolytic bacterial isolate *A. chroococcum* B4 ability in degradation of chicken feather keratin and production of keratinase activity was assessed in this study. *A. chroococcum* B4 was inoculated in liquid media containing chicken feathers as nutrient sole source. It was observed that *A. chroococcum* B4 successfully colonized and degraded chicken feathers starting from 2-days of incubation until 5 days of observation. Gupta and Singh5, and Fakhfakh et al.16 reported similar result in *Bacillus pumilus* A1. However, Jeong et al.10 showed that their isolate *Stenotrophomonas maltophilia* R13 took 6 days to completely degrade chicken. Feather degradation was observed using scanning electron microscopy. Visualization of the chicken feather degradation during cultivation was observed by scanning electron microscopy (SEM) (Fig. 1).

Bacterial ability in degrading feather was associated with two enzymes, serine protease and disulphide reductase as showed by *Stenotrophomonas* sp. D-1.17,18. As far as we know, no report of *A. chroococcum* was as keratinolytic bacteria. Microbial degradation of keratin releases peptides and amino acids including lysine, alanine, glycine, cysteine, valine, serine, and small quantities of tryptophan and methionine19, which can be used as N and C sources, and ultimately producing ammonium20.

SDS-PAGE and zymogram analysis of B4 keratinase

Molecular weight of keratinase was measured using SDS-PAGE method. The keratinase samples were from 23, 24, 25, and 35 fractions of Sephadex G-50 chromatography. SDS-PAGE analysis showed two major bands corresponding to fractions 24 and 25 (Fig. 2a). No protein bands were observed from other samples. SDS-PAGE result was reconfirmed by zymogram using keratin as a substrate dissolved in polyacrylamide gel. A single hydrolysis band of 24 and 25 fractions was formed in the gel (Fig. 2b). It was confirmed that molecular weight of B4 keratinase was 30 kDa.

In our study, we did not observe any presence of protein bands both in 70% ammonium sulphate precipitated and dialyzed crude keratinase. The phenomenon was due to the presence of ammonium salts which disrupted the keratinase activity of *A. chroococcum* B4 during partial purification. In the beginning of dialysis, majority of salts were removed but not all of them, as shown from fraction 23 th revealing a clear and thin band. As a result, fraction 24 and 25 th produced a thick band with the highest keratinase activity. The thick bands indicated a relatively high concentration of protein. The presence of non-keratinase protein bands was assumed to be other protease in our samples. Therefore, a zymogram

![Fig. 2. Visualization of SDS-PAGE (a) and zymography (b) of keratinase from *A. chroococcum* strain B4. Five samples of keratinase purification protocol were analysed: lane 1= ammonium sulfate (70%), lane 2= dialysis, lane 3= fraction (23 th), lane 4= fraction (24 th), lane 5= fraction (25 th), lane 6= fraction (35 th). M = Protein Marker, Color Prestained Protein Standard (BioLabs Inc, New England).](image-url)
analysis was performed to selectively estimate the molecular weight of our purified keratinase by using chicken feather as its substrate of hydrolysis.

Previous study showed that molecular weight of keratinases might vary between 18-240 kDa. In general, keratinases have molecular weight lower than 40 kDa, such as 22 kDa of B. subtilis, 28 kDa of B. licheniformis, 32.8 and 35.5 kDa of B. licheniformis YJ4, 36 kDa of S. maltophilia N4, 35.2 kDa of S. maltophilia DHJ. Larger keratinase was found in F. pennavorans with molecular weight of 130 kDa. The result of this study suggested that B4 keratinase showed to have similar molecular weight to those of Bacillus and Meiothermus.

Fig. 3. Keratinase activity during growth of A. chroococcum B4 on chicken feathers supplemented with different C- and N- sources. (a) C- source. (b) N-source. Values are the means ± SD of three replicates.

Fig. 4. Bacterial growth, keratinase activity and soluble protein release growth of A. chroococcum B4 on chicken feathers supplemented with C- and N- sources. Bacterial growth in medium supplemented with (a) 1% (w/v) sucrose (c) with 0.5% (w/v) peptone (b) Soluble protein and enzyme activity in medium supplemented with 1% (w/v) sucrose and (d) 0.5% (w/v) peptone.
Effect of C and N sources on keratinase production

Keratinase of *A. chroococcum* strain B4 was produced by optimization of submerged fermentation (SmF) as described previously using different C and N sources. Sucrose showed to increase keratinase production by 71.7 U/ml in 96 h of cultivation (Fig. 3a). On the other hand, monosaccharide such as glucose resulted to decrease keratinase production. Inhibition of protease biosynthesis by certain monosaccharides is likely correlated with catabolite repression mechanism. However, not all keratinase production by bacteria was inhibited by glucose. Keratinase production by *B. subtilis*, *B. licheniformis*, and *Stenotrophomonas* sp. D-1 were inhibited by glucose. Seemingly, glucose might stimulate synthesis keratinases in *B. pseudofirmus* AL-89 and *Bacillus* sp.

Various result of keratinase production was demonstrated due to different N sources. The addition of several nitrogen sources by 0.5% (w/v) resulted in an increase in keratinase synthesis by *A. chroococcum* B4 (Fig. 3b). High keratinase activity of 97.8 U/mL was observed when tryptone was added. Although the only source of C and N in controls were chicken feathers, the additional sources of C and N can help bacterial cell growth before the keratin is hydrolyzed. Having complex structure of keratins, exogenous nutrient source may contribute to keratinolytic bacteria earlier in feather degradation process. Consider that sucrose and tryptone are expensive, other C and N sources from other waste can be examined for increasing keratinase production.

Measurement of bacterial growth as optical density at 600 nm was carried out using spectrophotometer from 24 to 168 h (Fig. 4). The initial growth phase occurs after 24 h of cultivation, for addition both sucrose and tryptone. The logarithmic phase observed between 24 and 96 h of cultivation, while the stationary phase was seemingly not well defined. The bacterial cell growth was correlated with the keratinase activity and soluble protein produced. Data showed that maximum keratinase activity and soluble protein were at 96 h in sucrose (Fig. 4a-b) and 120 h in tryptone medium (Fig. 4c-d).

The increase of bacteria cell growth depends on the provision of growth substrate. More efficient production of primary metabolites was expected to occur in the logarithmic phase. Bacteria growth rate significantly affect the product during the fermentation process; the faster the growth rate, the less the fermentation process time, which then also leads to less production cost.

Effect of metal ions and inhibitors on keratinolytic activity of *Azotobacter chroococcum* B4

| Metal ions and Inhibitors | Relative activity (%) | 5 Mm | 10 Mm |
|---------------------------|-----------------------|------|-------|
| Control                   | 100                   | 100  |       |
| Ca^{2+}                   | 115 ± 0.9             | 117 ± 0.7 |
| Mg^{2+}                   | 112 ± 0.6             | 166 ± 1.0 |
| Ba^{2+}                   | 85 ± 1.2              | 110 ± 1.0 |
| Mn^{2+}                   | 109 ± 0.7             | 111 ± 0.8 |
| Co^{2+}                   | 51 ± 0.7              | 40 ± 0.8 |
| Na^{+}                    | 103 ± 0.9             | 113 ± 0.6 |
| K^{+}                     | 102 ± 1.0             | 112 ± 1.2 |
| Pepstatin A               | 51 ± 1.0              | 33 ± 1.5 |
| PMSF                      | 65 ± 0.9              | 31 ± 0.9 |
| DTT                       | 42 ± 0.7              | 22 ± 0.7 |
| EDTA                      | 52 ± 0.7              | 110 ± 0.7 |
| N-tosyl-L-lysine          | 72 ± 1.5              | 64 ± 0.9 |
| Bromoacetic acid          | 87 ± 1.5              | 35 ± 0.9 |
| Chymostatin               | 15 ± 0.9              | 14 ± 0.7 |
| Iodoacetic acid           | 51 ± 0.7              | 33 ± 1.0 |
| Benzamidine               | 57 ± 0.7              | 12 ± 0.9 |
| Soybeane trypsin inhibitor| 75 ± 0.9              | 66 ± 1.3 |
| 2-mercaptoethanol         | 75 ± 1.3              | 73 ± 0.9 |

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Effect of metal ions and inhibitors on keratinase B4

Metal ions have a significant role in the microbial metabolism, affecting the activity of many enzymes. The increase of keratinase B4 activity was observed when Ca^{2+}, Mg^{2+}, Mn^{2+}, Na^{+}, and K^{+} were added, with relative activities of 117%, 166%, 111%, 113%, 112%, respectively (Table 1).

However, Co^{2+} decreased the relative activity of keratinase B4 to 40%. In present report, Mg^{2+} can also be considered as significant factor in order to increase B4 keratinase activity, as also shown by keratinase of *Kocuria rosea*, *B. subtilis* and *Paracoccus* sp. However, metal ions may have different effect in different strains. For example, Ca^{2+} increased *S. brevicaulis* keratinase production, but decreased keratinase of *B. licheniformis* RG1.

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| 2-mercaptoethanol         | 75 ± 1.3              | 73 ± 0.9 |
It was known that various inhibitors decreased relative activity of keratinases. Inhibitor assays aim to determine keratinase type, considering that protease groups are very sensitive to presence of specific inhibitors. It has been reported previously that microbial keratinase was mostly related to serine proteases and metalloproteases, except that of yeast group that produced aspartic proteases. The keratinase B4 was inhibited by PMSF, benzamidine, chymostatin, which are well-known serine protease inhibitors. This indicated that B4 keratinase belonged to serine protease family. DTT, iodoacetic acid, and 2-mercaptoethanol also inhibited the enzyme, suggesting that cysteine residues were relevant for the activity of B4 keratinase as well.

CONCLUSIONS

Reported strain *A. chroococcum* B4 was able to degrade whole chicken feathers after 5 days of cultivation as evidenced by scanning electron microscopy. The analysis conducted by SDS-PAGE and zymogram showed that *A. chroococcum* B4 keratinase had a molecular weight of about 30 kDa. The keratinase activity increased after the addition of sucrose (1% w/v) as carbon source and tryptone (0.5% w/v) as nitrogen source during cultivation on feathers, reaching 97.8 U/mL. The enzyme activity enhanced by addition of Ca$^{2+}$, Mg$^{2+}$, Mn$^{2+}$, Na$^+$, and K$^+$, but it was inhibited by PMSF and other inhibitors indicating its belongingness to serine-protease group.

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CONFLICT OF INTEREST

The authors declare that there is no conflict of interest.

AUTHORS’ CONTRIBUTION

DS: concept of the main idea, developing the supporting theory, performing the experiment, writing of manuscript. EM: verifying the analytical methods. JM.: developing the theoretical formalism, performing the experiment. AZM: providing the laboratory necessity, supervising the findings of this project.

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ETHICS STATEMENT

This article does not contain any studies with human participants or animals performed by any of the authors.

DATA AVAILABILITY

All datasets generated and analyses are provided in this manuscript. The raw data may be requested upon acceptance by the authors.

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