Remote Plasma Atomic Layer Deposition of SiN$_x$ Using Cyclosilazane and H$_2$/N$_2$ Plasma

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Received: 5 August 2019; Accepted: 26 August 2019; Published: 28 August 2019

Abstract: Silicon nitride (SiN$_x$) thin films using 1,3-di-isopropylamino-2,4-dimethylcyclosilazane (CSN-2) and N$_2$ plasma were investigated. The growth rate of SiN$_x$ thin films was saturated in the range of 200–500 °C, yielding approximately 0.38 Å/cycle, and featuring a wide process window. The physical and chemical properties of the SiN$_x$ films were investigated as a function of deposition temperature. As temperature was increased, transmission electron microscopy (TEM) analysis confirmed that a conformal thin film was obtained. Also, we developed a three-step process in which the H$_2$ plasma step was introduced before the N$_2$ plasma step. In order to investigate the effect of H$_2$ plasma, we evaluated the growth rate, step coverage, and wet etch rate according to H$_2$ plasma exposure time (10–30 s). As a result, the side step coverage increased from 82% to 105% and the bottom step coverages increased from 90% to 110% in the narrow pattern. By increasing the H$_2$ plasma to 30 s, the wet etch rate was 32 Å/min, which is much lower than the case of only N$_2$ plasma (43 Å/min).

Keywords: atomic layer deposition; remote plasma; SiN$_x$ thin films

1. Introduction

Dielectric films, such as silicon nitride (SiN$_x$), are extensively studied as etch stop layers, gate dielectrics, stress liners, charge trap layers, and as spacer applications in front-end-of-line (FEOL) semiconductor wafer processing. The main applications utilize SiN$_x$ films as gate spacers in dynamic random access memory, logic devices, and the charge trap layer of vertical NAND flash devices [1,2]. These spacer films act as an oxygen or dopant out-diffusion barrier and control the source/drain doping profiles. They also act as a film to prevent etching damage during later processing. The requirements of a suitable SiN$_x$ spacer film include resistance to etching and high conformality [3–6]. However, it has been a challenge to develop new methods to ensure these requirements. Recently, gate spacer research has considered methods to control the dielectric constant by doping carbon into SiN$_x$ thin films to reduce the resistive-capacitive (RC) delay [7].

A variety of methods exist for depositing SiN$_x$ thin films, including low-pressure chemical vapor deposition (LPCVD), plasma-enhanced CVD (PECVD), and atomic layer deposition (ALD). Commonly used LPCVD is capable of producing highly conformal films at high temperature (approximately at 700 °C), with excellent etching properties and low hydrogen content [1,8,9]. However, the high deposition temperature exceeds the thermal budget of the gate spacer fabrication process. SiN$_x$ films using PECVD can be grown at low temperatures (≤400 °C); however, these films show low quality and poor step coverage [8–10].
As device feature size is miniaturized, the allowed thermal budget during processing decreases and film quality, such as conformality, must remain high [11]. The ALD method satisfies these requirements as a gate spacer fabrication technique because it provides precise thickness control, perfect step coverage, and high quality, even at low temperatures due to its self-limited reaction [12–15]. Thermal ALD induces reaction kinetics at the surface by heating the substrate or chamber, while plasma-enhanced ALD (PEALD) improves low-temperature reactivity by supplying additional energy from the plasma [10]. PEALD provides a variety of advantages for deposition of thin films compared to thermal ALD and other vapor phase deposition techniques. The high reactivity of the plasma generated during plasma-assisted ALD widens the choice of processing conditions and materials [10–13]. Also, the plasma radicals and ion fluxes are very high at the deposition surface in PEALD. Short plasma exposure times allow the entire surface to be deposited uniformly. However, the ion energy reaching the surface is so high that it can damage the surface. On the contrary, remote plasma ALD (RPALD) was developed to reduce plasma-induced damage on the surface. The reaction chamber and the plasma generation region are separated. There is much higher radical flux that contributes to deposition towards the substrate than direct plasma. In the case of PEALD, the gas temperature is easily changed in accordance with the substrate temperature change, which in turn affects the density of gaseous species and the generation of plasma species. The RPALD allows flexible control of plasma composition and properties at the substrate’s location, making it well-suited for process design [8,16–19]. The ALD of SiN$_x$, silicon chlorides (e.g., SiCl$_4$, SiH$_2$Cl$_2$, and Si$_2$Cl$_6$) have been studied as silicon precursors along with reactants such as NH$_3$, NH$_3$ plasma, or N$_2$H$_4$. The use of chlorosilane requires a high deposition temperature and causes detrimental effects such as particle formation, byproducts, and undesired chlorine incorporation. Therefore, chlorine-free precursors such as silane or aminosilane are in high demand [1,3,9,10,13].

PEALD lowers the substrate temperature due to the high reactivity of the plasma species, while less thermal energy is required at the surface reaction. Various precursors have been studied to deposit SiN$_x$ via PEALD, such as silane (SiH$_4$) [10], bis(tert-butylamino)silane (BTBAS) [3,4], di(sec-butylamino)silane (DSBAS) [4], trisilylamine (TSA) [1,8], trimethylsilane (TMS) [20], and bis(dimethylaminomethylsilyl)-trimethylsilylamine (DTDN2-H2) [9]. Coreactants such as NH$_3$ or N$_2$ plasma in combination with these silicon precursors were used. The PEALD of SiN$_x$ using aminosilane and NH$_3$ plasma leads to a reduced growth rate because the H- and NH$_x$-terminated surface is not undercoordinated, which in turn inhibits precursor adsorption. In contrast, N$_2$ plasma is able to generate reactive undercoordinated bare surface sites; PEALD using aminosilane and N$_2$ plasma reveals quite sensible growth rates [21–24].

In this study, SiN$_x$ thin films were deposited by RPALD using CSN-2 and N$_2$ plasma which were chosen as a precursor and a reactant gas, respectively (two-step). The liquid-type CSN-2 (provided in the DNF solution) used in this study has excellent thermal stability, high reactivity, and sufficient vapor pressure (4.04 torr at 60 °C). The process window and the physical and chemical properties of the film were evaluated using thin film analysis. A better quality RPALD SiN$_x$ film was produced and optimized by introduction of an intermediate H$_2$ plasma step between the CSN-2 and the N$_2$ plasma step (three-step). We carried out a comparative study between the two- and three-step processes at 350 °C to evaluate parameters such as step coverage by TEM analysis and wet etching characteristics to determine film suitability for gate spacer processing which requires a low thermal budget. Lee et al. performed research on SiN$_x$ thin films using CSN-2 with PEALD [1]. In this paper, as mentioned earlier, RPALD was used to investigate the physical properties of the SiN$_x$ films, such as atomic bonding states and step coverage. We used pure H$_2$ plasma as a reactant, which was not used in previous studies; in turn, the SiN$_x$ thin film and the ratio of the H$_2$ and N$_2$ gas mixture will be investigated later. Also, further study will be conducted to compare the difference in the physical properties between SiN$_x$ deposited by RPALD and PEALD under the influence of H$_2$ plasma reactant in the ALD process.
2. Materials and Methods

The Si wafer was a two-inch P-type Si (100) substrate for the deposition of SiNₓ, cleaned with diluted hydrofluoric acid for 2 min to remove native oxides. After cleaning, the substrate was loaded in the RPALD chamber. At first, SiNₓ films were deposited by RPALD with CSN-2 as the precursor and N₂ plasma as the reactant (two-step). Precursors were heated at 60 °C using a heating jacket to obtain sufficient vapor pressure and Ar (50 sccm) was used as the carrier gas in the line during the precursor dosing and purging step. The delivery lines were heated to 70 °C to prevent precursor condensation. N₂ plasma was generated by inductively coupled plasma (ICP) generated at a radio frequency of 13.56 MHz (Figure 1a). Deposition temperature was set in a wide range from 100 °C to 600 °C and the plasma power was fixed at 100 W. The recipe was selected as a standard which is composed of five stages (Figure 1b): CSN-2 dose time (3 s), precursor purge time (30 s), N₂ reactant gas injection time (4 s), N₂ plasma exposure time (20 s), and purge time (30 s). As shown in Figure 1c, we have developed the three-step process mentioned above, which adds H₂ gas injection (4 s) and an H₂ plasma exposure step (10, 20, and 30 s time variables) between the precursor purge and N₂ gas injection step. In the previous ALD cycle, the precursor delivery lines were purged with 700 sccm of Ar to remove byproducts and residual gas. A base pressure in the reactor chamber of ~10⁻⁶ torr was obtained using a turbo-molecular pump.

The thickness and refractive index of the deposited SiNₓ films were measured by spectroscopic ellipsometry (SE) using a Nano-View SE MG-1000 operated at an incident angle of 70° (1.5–5.0 eV). To investigate the surface morphology of SiNₓ thin films, atomic force microscope (AFM; Park Systems, XE-7) was used. The chemical compositions of the SiNₓ films were investigated with auger electron spectroscopy (AES). The chemical bonding state was determined by X-ray photoelectron spectroscopy (XPS) using a PHI 700Xi with Mg Kα X-ray source (E = 1.254 keV). Film wet etch rates were evaluated in a diluted HF solution (H₂O / HF = 100:1). Film thickness and step coverage were examined by transmission electron microscopy (TEM).

Figure 1. Schematic images of the (a) remote plasma atomic layer deposition (RPALD) system and process sequence (b) without H₂ plasma and (c) with H₂ plasma.
3. Results and Discussion

3.1. Process Window

The deposition rate of SiNₓ thin films was investigated as a function of precursor dosing time and plasma exposure time at 400 °C. The growth rate per cycle (GPC) increased and saturated at 0.38 Å/cycle, as CSN-2 dosing time increased to 5 s. An apparent saturation for GPC can also be investigated when the plasma exposure time was 20 s or longer. This GPC saturation indicates that the SiNₓ thin film RPALD process using CSN-2 and N₂ plasma is a self-limited reaction with no thermal decomposition of the precursor.

We investigated the growth rate of SiNₓ thin films at various temperatures (100–600 °C) as shown in Figure 2. The growth rate of the SiNₓ thin film was nearly constant at 0.38 Å/cycle in the deposition temperature range of 200–500 °C, known as the ALD process window. In the temperature regions below 200 °C and above 500 °C, the self-limited reaction is disturbed depending on process temperature. The former is due to precursor condensation on the substrate and the latter is due to thermal decomposition of the precursor; a “CVD-like” process [25–27]. As the deposition temperature increased, the refractive index of SiNₓ increased to 1.98 which is similar to 2.01 of the stoichiometric SiNₓ thin film, indicating that the quality of the film was improved. It can be also assumed that the film density is improved by increasing the refractive index [28].

![Figure 2. Effect of growth rate and refractive index as a function of deposition temperature at a plasma power of 100 W.](image)

3.2. Characteristics of SiNₓ Thin Films

AFM was used to analyze the roughness of the SiNₓ thin film. Figure 3 shows AFM images of SiNₓ thin film deposited at 250 °C, 350 °C, and 500 °C, respectively. Figure 3a has an RMS (root mean square) value of 0.077 nm, if SiNₓ thin film was deposited at 500 °C, the value of the SiNₓ film is 0.058 nm. Compared with the SiNₓ thin film deposited at low temperature, the roughness was slightly decreased as the process temperature increased. However, the width of the increase is so small that all thin films seem to have high surface quality. Even though the thin film is processed by various temperature conditions, good uniformity in the film can be observed.

To investigate the chemical compositions of the SiNₓ thin films, depth profiles were measured by AES, as shown in Figure 4. According to the AES analysis, there was an absolute decrease in carbon content with deposition temperature, which means that sufficient thermal energy was contributed to the reaction. Also, the films deposited at low temperature were vulnerable to oxidation due to low film density. SiNₓ thin films deposited below 250 °C contain a considerable amount of oxygen, which was reduced with increasing deposition temperature to 500 °C and measured to be less than 2%. The constant N and Si atomic contents indicate a uniform film stoichiometry throughout the entire film thickness. The ratio of N to Si was 1.34, which is nearly a stoichiometric Si₃N₄ film.
We investigated the growth rate of SiNx thin films at various deposition temperatures. Also, the films deposited at low temperature were vulnerable to oxidation due to low thermal energy. In Figure 2, the step coverage of the films improved with increasing deposition temperature. The step coverage of the SiNx thin film was investigated using trench-patterned wafers, as shown in Figure 3 (5 × 5 µm). As shown in Figure 4, Auger electron spectroscopy (AES) depth profiles of RPALD SiNx film deposited at 250 °C, 350 °C, and 500 °C. The chemical binding states in each film were investigated using XPS analysis. With the XPS scans of Si 2p and N 1s peaks, we obtained information regarding the bonding behavior of SiNx, as shown in Figure 5 (N 1s peak not shown). The XPS spectrum corresponding to the Si 2p with peak at 101.82 eV is the signature of the Si–N bond. The spectra corresponding to the N 1s content with peaks centered at 397.72 eV are the signature of the N–Si bond. The Si 2p spectra showed that all the samples had peaks at 101.82 eV. In the N 1s spectra, all the samples exhibited a peak at 397.72 eV [16]. In other words, the deposited samples were nearly stoichiometric Si3N4 and no difference in binding energy state was shown when the deposition temperature increased from 250 to 500 °C. As shown in Figure 5, the Si 2p peak deconvolution was performed with the Si–N binding energy corresponding to 101.82 eV and the Si–O binding energy corresponding to 103.27 eV. References to XPS and AES data confirmed that the oxygen content decreased sharply as the deposition temperature increased [16].

Figure 3. Atomic force microscopy (AFM) images (5 × 5 µm) of RPALD SiNx film deposited at (a) 250 °C, (b) 350 °C, and (c) 500 °C.

Figure 4. Auger electron spectroscopy (AES) depth profiles of RPALD SiNx film deposited at (a) 250 °C, (b) 350 °C, and (c) 500 °C.

Figure 5. Si 2p peak deconvolution of X-ray photoelectron spectroscopy (XPS) spectra for RPALD SiNx film deposited at (a) 250 °C, (b) 350 °C, and (c) 500 °C.
The step coverage of SiNₓ thin film was investigated using trench-patterned wafers, as shown in Figure 6. The aspect ratio of the trench patterned wafer is 2.7 with a top trench width of 31.2 nm in a narrow pattern. SiNₓ thin film thickness was measured at the top, side, and bottom of the trench with film conformality. The step coverage of the films improved with increasing deposition temperature. In the narrow pattern, the temperature condition of 250 °C yielded side and bottom coverages of 73% and 80%. As the deposition temperature increased to 500 °C, improved film conformality was obtained with side and bottom coverages of 81% and 87%.

![Figure 6. Transmission electron microscopy (TEM) images for step coverage of the RPALD SiNₓ film as a function of deposition temperature in a wide pattern at (a) 250 °C, (b) 350 °C, and (c) 500 °C and in a narrow pattern at (d) 250 °C, (e) 350 °C, and (f) 500 °C.](image)

The wet etch rate (WER) test was performed to investigate SiNₓ etching properties. A diluted HF solution (1:100) was used. The LPCVD of Si₃N₄ film at 730 °C was referenced for the WER test. As shown in Table 1, the wet etch rate decreased with increasing deposition temperature. In the above-mentioned Figure 2, as the deposition temperature increases, the film density also improved due to the increase of the refractive index [28,29]. It is understandable that the wet etch rate decreases with increasing deposition temperature, considering the fact that the film density is inversely proportional to WER [17].

| Temp. (°C) | 250 | 350 | 500 |
|-----------|-----|-----|-----|
| Wet Etch Rate (Å/min) | 57  | 43  | 35  |

![Table 1. Wet etch rate of the RPALD SiNₓ film as a function of deposition temperature.](image)

The physical properties of the SiNₓ film based on the CSN-2 precursor have been studied. Thin films with good step coverage and excellent etching properties were obtained at 500 °C. In order to improve the progress made through the previously described two-step process (CSN-2/purge/N₂ plasma/purge), an H₂ plasma step was added between the CSN-2 and N₂ plasma step to improve the characteristics of the thin film used as a gate spacer. In logic devices, the actual gate spacer process proceeds at temperatures below 400 °C to prevent implant out-diffusion and unwanted metal oxidation in the high-k metal gate (HKMG). Thus, we performed a comparative study between the two-step and three-step processes at 350 °C.
N\textsubscript{2} plasma not only removes the ligand of the precursor, but also facilitates the chemisorption of the silicon precursor. However, the N radicals have an exceptionally short lifetime compared to H radicals or O radicals. Due to this short lifetime, recombination loss occurs and the ligand is not completely removed from the precursor, such that the step coverage and quality of the thin film deteriorates. By introducing the H\textsubscript{2} plasma before the N\textsubscript{2} plasma, efficient ligand removal from the surface can be achieved by H radicals with a long lifetime. The SiN\textsubscript{x} films with an excellent step coverage and wet etch rate were obtained [21–24].

As shown in Figure 7a, as the H\textsubscript{2} plasma exposure time increases in the three-step process, the deposition rate decreases because the influence of the H\textsubscript{2} plasma forms an NH\textsubscript{2} or SiH group, which make reactions with the precursor difficult. In contrast, the deposition rate improved by fixing the H\textsubscript{2} plasma and increasing the N\textsubscript{2} plasma time as seen in Figure 7b. This is because the N\textsubscript{2} plasma regenerates the reaction sites on the surface where the -H or -NH\textsubscript{2} group is terminal and forms an undercoordinated surface. Therefore, the precursor easily reacts with the surface [16].

As shown in Figure 8, TEM images show step coverage with various H\textsubscript{2} plasma exposure times. Compared with the case of As-dep films, the step coverage improves with increasing H\textsubscript{2} plasma exposure time. In general, we can observe that the step coverage is greatly improved at the bottom/top compared to the side/top. In the narrow pattern, the side step coverage increased from 82% to 105% and the bottom step coverage increased from 90% to 110%. The narrow pattern is shown in Figure 8d–f. When the H\textsubscript{2} plasma exposure time is 20 s, conformality was as good as the thin film proceeded at 500 °C. In Table 2, the WER was investigated at the same condition. As observed, the longer the H\textsubscript{2} plasma exposure time, the lower the etch rate, which decreased from 43 to 32 Å/min.

Table 2. Wet etch rate as a function of H\textsubscript{2} plasma exposure time in the three-step process.

| H\textsubscript{2} Plasma Time (s) | 0  | 10 | 20 | 30 |
|----------------------------------|----|----|----|----|
| WER (Å/min)                     | 43 | 39 | 34 | 32 |

It was confirmed that the step coverage and wet etching characteristics were improved with H\textsubscript{2} plasma time by applying the low temperature process, which is necessary for the deposition of the gate spacer. Through process improvement, we were able to concentrate on the physical properties of the SiN\textsubscript{x} thin film. It was found that a high quality thin film could be obtained for the gate spacer depending on how the unique characteristics of the plasma reactant are utilized in the ALD process.
Figure 8. Cross sectional TEM images for step coverage of the RPALD SiNₓ film as a function of H₂ plasma exposure time in a wide pattern for (a) 10 s, (b) 20 s, and (c) 30 sec and in a narrow pattern for (d) 10 s, (e) 20 s, and (f) 30 s. All the samples were deposited at 350 °C.

4. Conclusions

In this work, physical and chemical properties of SiNₓ as a gate spacer were evaluated. The process environment was implemented in a remote plasma ALD system designed to overcome the limitations of the high thermal budget of thermal ALD and to minimize the thin film damage that can occur in direct plasma ALD [19]. The atomic concentration and chemical binding state were confirmed by deposition of SiNₓ with CSN-2 and N₂ plasma. The fundamental properties of step coverage and wet etch rate were analyzed. The H₂ plasma step was introduced before the N₂ plasma step, which is a three-step process resulting in a reduction of growth rate. Many studies using NH₃ plasma as a reactant have been performed. In our research, the H₂ plasma and N₂ plasma were used as reactants, respectively, instead of NH₃ plasma. We compared the effect of the two-step process (only N₂ plasma) with the three-step process at low temperature (≤400 °C) [22]. According to the results of the study for the narrow pattern, the longer H₂ plasma exposure time increased the side step coverage from 81% to 105%, and bottom step coverage increased from 90% to 110%. In addition, the wet etch rate was reduced from 43.2 to 32.4 Å/min. Based on the results of this study, it was confirmed that the step coverage and the wet etching characteristics of the gate spacer thin film can be appropriately controlled by the optimization process.

Author Contributions: Data curation, H.C. (Hyeongsu Choi), H.P., C.J., S.S., H.Y., Y.K., J.-W.K., K.K., Y.C., and S.P.; Original draft preparation, H.C. (Haewon Cho); writing—review and editing, N.L.; Supervision, H.J.

Funding: This work was supported by the Nano Material Technology Development Program (2014M3A7B4049367) through the National Research Foundation (NRF) of Korea funded by the Ministry of Science and ICT (MSIT), Korea. This work is supported by the Samsung Electronics’ University R&D program [Study of high step coverage ONO thin film for CTF device].

Conflicts of Interest: The authors declare no conflict of interest.
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