Inoculation with *Aspergillus aculeatus* Alters the Performance of Perennial Ryegrass under Phosphorus Deficiency

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**ABSTRACT.** Phosphorus (P) is an essential nutrient element that is necessary for plant growth and development. However, most of the P exists in insoluble form. *Aspergillus aculeatus* has been reported to be able to solubilize insoluble forms of P. Here, to investigate the P-solubilizing effect of *A. aculeatus* on the performance of perennial ryegrass (*Lolium perenne*) under P-deficiency stress, we created four treatment groups: control [i.e., no Ca₃(PO₄)₂ or *A. aculeatus*], *A. aculeatus* only (F), Ca₃(PO₄)₂ and Ca₃(PO₄)₂D*A. aculeatus* [Ca₃(PO₄)₂ + F] treatment, and Ca₃(PO₄)₂ at concentrations of 0 and 3 g per pot (0.5 kg substrate per pot). In our results, the liquid medium inoculated with *A. aculeatus* exhibited enhanced soluble P and organic acid content (tartaric acid, citric acid, and aminoacetic acid) accompanied with lower pH, compared with the noninoculated regimen. Furthermore, *A. aculeatus* also played a primary role in increasing the soluble P content of substrate (1 sawdust: 3 sand), the growth rate, turf quality, and photosynthetic capacity of the plant exposed to Ca₃(PO₄)₂ + F treatment, compared with other groups. Finally, in perennial ryegrass leaves, there was a dramatic increase in the valine, serine, tyrosine, and proline contents, and a remarkable decline in the glutamic acid, succinic acid, citric acid, and fumaric acid contents in the Ca₃(PO₄)₂ + F regimen, compared with other groups. Overall, our results suggested that *A. aculeatus* may play a crucial role in the process of solubilizing Ca₃(PO₄)₂ and modulating perennial ryegrass growth under P-deficiency stress.

Phosphorus is one of the major macronutrients that are essential for plant performance, whose deficiency constrains plant growth and crop productivity (Taktek et al., 2015). Naturally, P exists in organic and inorganic P forms. However, plants absorb only inorganic P, which due to its existence in an insoluble form has a limited abundance in the soil (Dobelaere et al., 2003; Richardson, 2001). Therefore, P deficiency in the soil is a critical problem for plant growth and agricultural yield (Taktek et al., 2015).

To cope with and overcome such problems, the application of chemical phosphate fertilizers to P-deficient soils is an inevitable strategy to mitigate P deficiency (Richardson, 2001). Unfortunately, after application, a large proportion of these P fertilizers are easily and rapidly converted to insoluble P because of complex reactions with other cations such as Fe²⁺/Fe³⁺, Al³⁺, and Ca²⁺ in acidic and alkaline soils (Chacon et al., 2006; Khan et al., 2007; Richardson, 2001). Hence, only a fraction of the P fertilizers introduced into the soil can be absorbed and used by plants (Liu et al., 2014).

To comprehensively address the detrimental effects of P deficiency on plant growth, some viable and extensive investigations have reported that soil microbes can solubilize insoluble phosphates into a soluble form by secreting phosphatases and organic acids, and thereby enhancing the availability of P to plants (Richardson et al., 2009; Zhu, 2002). For example, *Bacillus* and *Pseudomonas* bacteria, as well as strains of *Penicillium* and *Fusarium* fungi, are common P solubilizers (Whitelaw, 2000). In previous research, it was reported that phosphate-solubilizing bacteria mobilize P from the soil that...
contained organic and inorganic P; thereby, it could be considered as suitable biofertilizers to promote nutrition absorption by plants (Calvo et al., 2014). Furthermore, phosphate-solubilizing bacteria have been reported to play a pivotal role in the solubilization and mobilization of P, which enhanced soil fertility (Dobbelaere et al., 2003). Also, it was reported that naturally most terrestrial plants acquired phosphorus sources from the soil due to symbiotic mycorrhizal fungi (Smith and Read, 1997). Furthermore, the application of arbuscular mycorrhizal fungi accelerated plant growth and doubled the absorption of P under low or medium levels of P in soil (Weber et al., 1992). Therefore, according to the preceding observations, it was concluded that part of chemical fertilizers can be replaced by biofertilizers that have a better potential to counter P-deficient stress (Velázquez et al., 2010).

In this study, *A. aculeatus* was isolated from the rhizosphere of bermudagrass (*Cynodon dactylon*) by Xie et al. (2014a). It was confirmed that the fungus could colonize plant roots and secrete Indole-3-acetic acid, siderophores, and ACC deaminase, thereby alleviating the damage induced by adverse conditions to plants and promoting plant growth and yield (Li et al., 2017; Xie et al., 2017). In addition, *A. aculeatus* can be easily grown in axenic cultures allowing for further vast propagation due to the lack of specificity. Furthermore, an earlier study demonstrated that *A. aculeatus* had the ability to solubilize natural forms of P (Narsian and Patel, 2000) and solubilize soluble potassium (K) (X.N. Li, unpublished data), which promoted the absorption of P and K nutrient elements by plants. Furthermore, *A. aculeatus* can alleviate the damage caused by stress, such as salt stress (Li et al., 2017; Xie et al., 2017), cadmium stress (Xie et al., 2014a, 2018), high temperature stress (X.N. Li, unpublished data), and drought stress (X.N. Li, unpublished data) in ameliorating the performance of perennial ryegrass exposed to P deficiency is still unexplored. Therefore, *A. aculeatus* might be considered to be a kind of a biofertilizer when perennial ryegrass is exposed to P-deficiency stress. Perennial ryegrass, with high nutritive values and persistence, is applied extensively as an important forage grass species (Wilkins, 1991). In addition, it is a cool-season turfgrass that is widely used in golf courses, parks, and other places because of its fast establishment, and superior tillering and regeneration ability (Altpeter, 2006; Xiong et al., 2007).

Therefore, the aim of the present study was to explore the mechanism of *A. aculeatus* solubilizing insoluble P and its subsequent growth-promoting effect on plants grown in P-impoverysoil. Subsequently, we conducted two experiments to evaluate the fungal functions: 1) the P-solubilizing capacity of *A. aculeatus* in a liquid medium, and 2) the effect of *A. aculeatus* on the growth of perennial ryegrass under P deficiency. To perform the two experiments, we measured some crucial indicators, such as pH, soluble P and organic acid of the medium, soluble P of substrate soil, and plant growth indicators.

**Materials and Methods**

**Expt. 1: The phosphate-solubilizing capacity of *A. aculeatus* in a liquid medium**

**Fungal cultures.** The fungi *A. aculeatus*, belonging to the genus *Aspergillus* used in this experiment, were screened and identified by Xie et al. (2014a), which was isolated from the rhizosphere of bermudagrass in Hunan province, central China. Initially, the fungi were removed from a freezer at –80 °C and activated in the National Botanical Research Institute’s phosphate growth agar (NBRIP) medium including of 0.5 g (NH₄)₂SO₄, 0.3 g KCl, 0.3 g NaCl, 0.3 g MgSO₄·7H₂O, 0.03 g FeSO₄·7 H₂O, 0.03 g MnSO₄·H₂O, 10 g glucose, 20 g agar, diluted with distilled water to 1 L, which was kept in a growth chamber at 30 °C for 7 d (30 May to 6 June 2017).

**Ca₃(PO₄)₂ treatment and experimental design.** For Ca₃(PO₄)₂ treatment, 0, 3, and 5 g L⁻¹ Ca₃(PO₄)₂ were added to the liquid NBRIP medium containing 0.5 g (NH₄)₂SO₄, 0.3 g KCl, 0.3 g NaCl, 0.3 g MgSO₄·7H₂O, 0.03 g FeSO₄·7H₂O, 0.03 g MnSO₄·H₂O, 10 g glucose, with 1 L of distilled water. Subsequently, the liquid NBRIP medium was poured into a 250-mL erlenmeyer flask filled to 150 mL and sterilized in an autoclave for 20 min at 121 °C. In addition, the mycelium was gently scraped into the sterile water from the surface of the NBRIP agar medium on 8 June 2017, and then the fungus suspension (optical density at 600 nm = 0.65) was prepared. The liquid medium was divided into two groups: one was inoculated with 6 mL fungus suspension per 150 mL liquid medium and the other was left uninoculated. Finally, all flasks were placed on a shaker at 200 g, at 30 °C, and the supernatant was collected to measure the pH, soluble P content on days 0, 3, and 6 (8, 11, and 14 June 2017, respectively); and organic acid content in the medium on days 3 and 6.

**Measurements**

**Determination of pH and soluble P of medium.** After inoculation for 0, 3, and 6 d, the medium was centrifuged at 10,000 g for 5 min, and then the supernatant was collected. The pH of the medium was recorded using a pH meter equipped with a glass electrode. The content of soluble phosphate was determined by an automatic chemical analyzer (EasyChem Plus; Systea, Anagni, Italy) based on the molybdate blue method (Murphy and Riley, 1962).

**Organic acid of medium.** For the organic acids content analysis, a high-performance liquid chromatography (HPLC) system (Accela 1250; Thermo Fisher Scientific, Waltham, MA) was used on 15 June 2017 (Shui and Leong, 2002). The medium was centrifuged at 10,000 g, for 5 min when inoculated for 0, 3, and 6 d, and then the supernatant was collected. The supernatant was filtered through a 0.22-μm membrane filter and then 0.5 mL of filtrates was injected into 2-mL HPLC vials for HPLC determination. The separation was carried out on a 250 × 4.6-mm, 5-μm column (Diamonsil C18; Dikma Technologies, Foothill Ranch, CA). Detection was operated at 214-nm wavelength with 0.1 mmol L⁻¹ KH₂PO₄ (pH = 2.6) as mobile phase passed through a 0.45-μm membrane filter. The C18 column was operated with 0.5 mL min⁻¹ flow rate at 30 °C of column temperature and the sample injection volume was 20 μL. Every analysis was replicated three times.

**Expt. 2: The effect of *A. aculeatus* on the growth of perennial ryegrass under P deficiency**

**Preparation of the growth substances.** The growth substrate used in this study consisted of a mixture of sawdust (P-free with 3–6 mm particles) and sand [1 sawdust: 3 sand (v/v), pH = 6.5], which was prepared on 20 June 2017. All the sand mixes were sterilized in an autoclave for 1 h at 127 °C and then were dispensed into plastic pots (9.5 cm diameter and 18 cm deep; 0.5 kg mixture per pot). All the plastic pots were
drilled at the bottom to enhance the substrate ventilation and drainage and were sterilized by ultraviolet (Ultraviolet Ray; Philips, Amsterdam, The Netherlands) for 1 h. Subsequently, the growth substances were divided into two groups, namely Ca₃(PO₄)₂ and Ca₃(PO₄)₂ + A. aculeatus [Ca₃(PO₄)₂ + F].

The A. aculeatus isolate used in this study was activated with NBRIP agar medium for 7 d and then was inoculated in 250-mL erlenmeyer flasks filled with 150 mL NBRIP liquid medium. All flasks were kept on the shaker at 200 rpm, 30 °C for 7 weeks to allow for establishing colony-forming units/mL) were inoculated into the pot and thoroughly mixed into the substrate on 23 June 2017; and then substrates were cultivated in a growth chamber at 30 °C for 48 h before the seeds were sowed.

**Plant materials and growth conditions.** The perennial ryegrass ‘Lark’ seeds (0.5 g per pot) were surface-sterilized with 70% ethanol (5 min) and then with 0.1% HgCl₂ (5 min), and then washed five times using sterile water. Subsequently, the seeds were sown evenly in plastic pots (9.5 cm diameter, 18 cm deep) filled with the pre-prepared mixture of sand and sawdust (as described previously) and covered by a 0.5-cm layer of sand on 26 June 2017. After germination, the materials were watered daily and fertilized twice weekly with half-strength Hoagland nutrient solution (in the absence of phosphorus source) including KNO₃ (2.5 mM), Ca(NO₃)₂ (2.5 mM), MgSO₄·7H₂O (1 mM), H₂BO₃ (1.43 mg·L⁻¹), ZnSO₄·7H₂O (0.11 mg·L⁻¹), CuSO₄·5 H₂O (0.04 mg·L⁻¹), MnCl₂·4 H₂O (0.91 mg·L⁻¹), H₂MoO₄ (0.05 mg·L⁻¹), Fe–EDTA (0.04 mM). All grasses were maintained in a greenhouse with natural illumination for 7 weeks to allow for establishing the leaves and roots of the plant with a daily temperature of 21 ± 3/18 ± 3 °C (day/night) and 14-h photoperiod (from 26 June to 14 Aug. 2017).

**Experimental design and sampling.** All materials were divided into four groups: control (CK [no Ca₃(PO₄)₂ or A. aculeatus in medium]), A. aculeatus only (F), Ca₃(PO₄)₂, and Ca₃(PO₄)₂ + A. aculeatus [Ca₃(PO₄)₂ + F] treatment, and Ca₃(PO₄)₂ concentrations were 0 and 3 g per pot (0.5 kg substrate per pot). After 7 weeks of the establishment, the shoot lengths were measured to assess the relative growth rate and subsequently the leaf samples were harvested for accessing other parameters on 14 Aug. 2017. During harvest, the medium was collected for measuring the PH and soluble P of medium. The Ca₃(PO₄)₂ treatment and concentration were randomized in a complete block design with five replicates.

**Measurements**

**Determination of soluble P in the growth substrate.** The 2.5 g of crushed and screened substrate samples were placed into a 250-mL erlenmeyer flask, in which 50 mL of 0.5 M NaHCO₃ extractant and 5 g non-P active carbon were added; and then the mixture was shaken on a rotary shaker for 30 min at 250 g. Subsequently, 6 mL of 14% H₂SO₄ was added to flasks to escape the CO₂ and then the mixture was filtered through a qualitative filter paper for further measurement. The content of soluble phosphate was measured with an automatic chemical analyzer (EasyChem Plus) based on molybdate blue method (Murphy and Riley, 1962) on 20 Aug. 2017.

**Determination of pH in the growth substrate.** Medium pH was evaluated by immersing a glass electrode with medium-water suspensions that were whisked with a glass rod for 2 min and kept for 30 min at room temperature on 21 Aug. 2017.

**Relative growth rate.** The relative growth rate of shoots (RGR) was estimated by measuring the turf vertical average height above the growth substances at the end of the experiment on 14 Aug. 2017. The final height was recorded as Hₜ at the end of the experiment. The duration of the experiment was recorded as Δt. The RGR of the materials was calculated according to the equation: \[ RGR = \frac{H_t}{\Delta t}. \]

**Turf quality estimation**

Turf quality was evaluated visually based on a scale of 0 to 9 on a weekly basis according to turfgrass color on 14 Aug. 2017, where 0 = yellow, withered, thin, and dead; 6 = minimum acceptable level based on density, turf color, and uniformity; and 9 = green, dense, and uniform (Turgeon, 1991).

**Quantification of chlorophyll a fluorescence transient**

Chlorophyll a fluorescence transient (OJIP transient) can be used to measure the photosynthetic processes efficiency and kinetics that are involved in photosystem II (PSII) (Chen et al., 2013). [About OJIP curve: F₀: minimal reliable recorded fluorescence, at 20 μs with the pulse-amplitude modulation (PAM) fluorometer; F₁: fluorescence intensity at the J-step (2 ms) of OJIP; F₂: Fluorescence intensity at the I-step (30 ms) of OJIP; Fₐ: maximal recorded fluorescence intensity, at the peak P of OJIP.] OJIP transient was determined on 14 Aug. 2017 according to the method described by Chen et al. (2013) using a pulse-amplitude modulation fluorometer (PAM 2500; Heinz Walz, Efelfrich, Germany) with a high time resolution of 10 μs. At the end of the experiment, the fourth fully expanded leaves were placed in a dark room for 30 min dark adaptation to close all reaction centers of PS II and obtain the maximal fluorescence yield. After dark adaptation, the OJIP curves were evaluated with a red light of 3000 μmol·m⁻²·s⁻¹ to acquire the maximum fluorescence intensity (Fₐ). The chlorophyll a fluorescence emission was induced by the strong light pulses and then was determined and digitized between 10 μs and 320 ms. The OJIP curve was analyzed by the JIP test as described in Table 1.

**Metabolite extraction and derivatization**

For the metabolite assay, fully expanded perennial ryegrass leaves (≈0.3 g) were harvested on 25 Aug. 2017 and stored in the refrigerator at −80 °C until further analysis. The metabolite extraction and sample derivatization were extracted according to the protocol as described previously (Xie et al., 2014b). The frozen plant samples were ground into a fine powder in liquid nitrogen with prechilled mortar and pestle, then the powder was transferred into a 2-mL centrifuge tube containing 4.2 mL of 80% (v/v) aqueous methanol. Subsequently, the tubes were shaken for 2 h at 200 rpm, ambient temperature, and then 60 μL of ribitol (2 mg·mL⁻¹) was added into the solution as internal standard. After that, the solution was heated in a water bath at 70 °C for 15 min and centrifuged for 15 min at 12,000 gₚ and then the supernatant was transferred into a new 10-mL centrifuge tube.
containing 4.5 mL of deionized water and 2.25 mL of chloroform. The mixture solution was vortexed fully for 15 s and centrifuged at 10,000 g, for 10 min. The supernatant (i.e., polar phase), 0.3 mL, was transferred into 2-mL HPLC vials and then was dried using a centrifugal concentrator at 900 g, overnight. The dried polar phase was derivatized with 80 mL of methoxymine hydrochloride (20 mg·mL⁻¹) dissolved in pyridine at 30 °C for 2 h and was trimethylsilylated with 50 μL N-methyl-N-trimethylsilyltrifluoroacetamide (MSTFA) for 2 h at 30 °C. The reagents used in our research were purchased from Sigma-Aldrich (Poole, UK).

**Gas chromatography–mass spectrometry (GC-MS) analysis**

The metabolites were measured using a GC-MS (7890A/5975C; Agilent Technologies, Palo Alto, CA) analysis based on the protocol of Xie et al. (2014b). For GC-MS, the derivatization samples (1 μL) were injected into a 30 m × 0.25 mm × 0.25-μm capillary column (DB-5MS; Agilent Technologies). The procedure was performed as follows: the inlet temperature was set at 280 °C and 5 min solvent delay. Subsequently, the initial GC oven temperature was set at 70 °C; 1 min after injection, the temperature of the GC oven was increased to 280 °C with 5 °C per min, and then kept at 280 °C for 10 min. The injection temperature was set at 280 °C, while the ion source temperature was adjusted to 230 °C. Helium was used as the carrier gas with a constant flow rate set at 1 mL per min. The measurement was recorded at two scans per s with 70 eV of electron impact ionization in a full scan mode (m/z range 30–650).

Metabolite data processing and analysis

The metabolites were processed and analyzed with software (MSD Productivity Chemstation; Agilent) by the retention time coupled with available compound libraries of National Institute of Standards and Technology (NIST 11; NIST, Gaithersburg, MD). Relative quantification of the metabolite was calculated according to the internal standard (ribitol).

Statistical analysis

Statistical analysis was performed using the program SPSS (version 16.0; IBM Corp., Armonk, NY). Statistical differences (P < 0.05) were estimated using the Student-Newman-Keuls test combining one-way analysis of variance.

### Results

The phosphate-solubilizing capacity of *A. aculeatus* in the liquid medium

The *A. aculeatus* exhibited strong phosphate-solubilizing characteristics. As shown in Fig. 1A–C, soluble P content was 12.7- or 10.2-fold higher in the Ca₃(PO₄)₂ + F group than in the Ca₃(PO₄)₂ group, when the concentration of Ca₃(PO₄)₂ was 3 or 5 g on the 6 d. As expected, in Fig. 2A–C, the pH of the liquid medium was remarkably decreased by *A. aculeatus* when compared with the Ca₃(PO₄)₂ group at days 3 and 6, regardless of Ca₃(PO₄)₂ concentration. When the concentration of Ca₃(PO₄)₂ was 0, 3, and 5 g, the pH of the liquid medium decreased by 2.90-, 1.98-, and 1.65-fold, respectively, in the Ca₃(PO₄)₂ + F group than in the Ca₃(PO₄)₂ group on day 6.

A decline in the pH of the liquid medium was accompanied by the production of organic acids by *A. aculeatus*. As observed in Table 1, when Ca₃(PO₄)₂ was absent in the liquid medium, the fungus also could produce organic acids (tartaric acid, malic acid, citric acid, and aminoacetic acid). In addition, when the concentration of Ca₃(PO₄)₂ was 3 g, the content of tartaric acid, citric acid, and aminoacetic acid in the liquid medium was 1.51-, 4.49-, and 2.14-fold higher on day 6 than day 3. Simultaneously, when the concentration of Ca₃(PO₄)₂ was 5 g, the content of tartaric acid and aminoacetic acid was 2.16- and 1.93-fold higher on day 6 than day 3.

**PLANT GROWTH PERFORMANCE.** The result is shown in Fig. 3, where *A. aculeatus* dramatically increased plant turf quality and relative growth rate, regardless of Ca₃(PO₄)₂ application. In addition, the turf quality was 1.30-fold higher in the Ca₃(PO₄)₂ + F regimen than in the Ca₃(PO₄)₂ regimen. Simultaneously, the relative growth rate had a 1.21-fold increase in the Ca₃(PO₄)₂ + F regimen than in the Ca₃(PO₄)₂ regimen (Fig. 3).

**Soluble P and pH of the medium.** *A. aculeatus* could solubilize Ca₃(PO₄)₂ into soluble phosphate in the medium. As exhibited in Fig. 4, in the Ca₃(PO₄)₂ + F group, the soluble P accumulation increased by 1.77-, 2.77-, and 2.87-fold, respectively, with the Ca₃(PO₄)₂ group, F group, and CK. Simultaneously, the pH of medium was 1.23-, 1.23-, and 1.07-fold higher after inoculation of fungi in the Ca₃(PO₄)₂ + F group compared with the control, F, and Ca₃(PO₄)₂ groups.

### Table 1. Parameters for analysis of tartaric, malic, citric, and aminoacetic acids in liquid medium via high-performance liquid chromatography on days 3 and 6 after the addition of *Aspergillus aculeatus* (F).

| Time | Treatment [g·L⁻¹ Ca₃(PO₄)₂] | Tartaric acid [mean ± SD (mg·mL⁻¹)] | Malic acid [mean ± SD (mg·mL⁻¹)] | Citric acid [mean ± SD (mg·mL⁻¹)] | Aminoacetic acid [mean ± SD (mg·mL⁻¹)] |
|------|-----------------------------|-----------------------------------|----------------------------------|-----------------------------------|-------------------------------------|
| Day 3 | 0 + F                        | 0.571 ± 0.02 a                     | 0.226 ± 0.11 a                   | 0.025 ± 0.00 a                    | 0.037 ± 0.02 a                      |
|      | 3 + F                        | 0.508 ± 0.02 b                     | 0.185 ± 0.01 a                   | 0.063 ± 0.03 a                    | 0.036 ± 0.01 a                      |
|      | 5 + F                        | 0.510 ± 0.01 b                     | 0.180 ± 0.04 a                   | UD                                | 0.059 ± 0.01 a                      |
| Day 6 | 0 + F                        | 0.770 ± 0.04 b*                    | UD                               | 0.055 ± 0.00 b*                   | 0.080 ± 0.01 a*                     |
|      | 3 + F                        | 0.768 ± 0.08 b*                    | UD                               | 0.283 ± 0.11 ab*                  | 0.077 ± 0.03 a                      |
|      | 5 + F                        | 1.101 ± 0.21 a*                    | UD                               | 0.466 ± 0.17 a                    | 0.114 ± 0.012 a*                    |

* Treatments consisted of 0, 3, and 5 g·L⁻¹ Ca₃(PO₄)₂ added to a liquid medium containing 0.5 g (NH₄)₂SO₄, 0.3 g KCl, 0.3 g NaCl, 0.3 g MgSO₄·7H₂O, 0.03 g FeSO₄·7 H₂O, 0.03 g MnSO₄·H₂O, and 10 g glucose, in 1 L distilled water. The liquid medium was divided into two groups: one was inoculated with 6 mL fungal suspension per 150 mL liquid medium and the other was left uninoculated for subsequent organic acid determination.

* Different letters indicate significant differences under three different Ca₃(PO₄)₂ concentration on days 3 or 6.

* Significant differences between days 3 and 6 with the same Ca₃(PO₄)₂ concentrations based on Student-Newman-Keuls test combining one-way analysis of variance (P < 0.05).

UD = undeterminable.
PHOTOSYNTHETIC EFFICIENCY. A combination of *A. aculeatus* and Ca$_3$(PO$_4$)$_2$ treatment notably affected the OJIP fluorescence transient curves of the plant leaves (Fig. 5). According to the observations, the OJIP transient curve in the Ca$_3$(PO$_4$)$_2$ + F group was higher than other groups (Fig. 5 combined with Table 2). To explore the OJIP fluorescence transient curves, the JIP test was applied. The basic fluorescence values were extracted from OJIP curves to analyze the structural and functional parameters. As shown in Table 2, basic parameters, including $F_J$, $F_I$, and $F_P$, displayed an upregulated trend when
plants were exposed to Ca$_3$(PO$_4$)$_2$ + F treatment, compared with other treatments.

The parameters of quantum yields and efficiencies, such as φE$_0$ and ΨE$_0$, were 1.13- and 1.15-higher in the Ca$_3$(PO$_4$)$_2$ + F regimen than Ca$_3$(PO$_4$)$_2$ counterpart, respectively. Moreover, the specific energy fluxes, such as ET$_0$/reaction center (RC) and RE$_0$/RC, showed 1.10-fold and 1.43-fold increase in the Ca$_3$(PO$_4$)$_2$ + F group, compared with the Ca$_3$(PO$_4$)$_2$ group. In addition, performance index on absorption basis (PI$_{ABS}$) and PI$_{total}$ played an important role in displaying the PSII overall activity. In the Ca$_3$(PO$_4$)$_2$ + F group, performance index PI$_{ABS}$ had a 1.28-, 1.27-, and 1.28-fold increase respectively, compared with the Ca$_3$(PO$_4$)$_2$ group, F group, and CK. The PI$_{total}$ was 1.36-, 1.5-, and 1.5-fold higher, respectively, in the Ca$_3$(PO$_4$)$_2$ + F group than the Ca$_3$(PO$_4$)$_2$ group, F group, and CK (Table 2).

Metabolite homeostasis

To investigate the alteration of metabolites as triggered by A. aculeatus when the plants were subjected to P-deficiency stress, a GC-MS was applied. There were 26 metabolites, including eight amino acids, eight organic acids, six sugars, three fatty acids, and myo-inositol (Table 3).

As described in Fig. 6, the amino acids, such as valine, serine, and glutamic acid had a 1.36-, 1.16-, and 1.17-fold higher content in Ca$_3$(PO$_4$)$_2$ + F group than Ca$_3$(PO$_4$)$_2$ only group. By contrast, the organic acids, such as citric acids, fumaric acids, and succinic acids content were 0.78-, 0.82-, and 0.94-fold lower in Ca$_3$(PO$_4$)$_2$ + F regime than Ca$_3$(PO$_4$)$_2$ only group.

Discussion

In this study, assessment of the A. aculeatus strain’s characteristic of P-solubilizing showed that the fungus had the capacity to solubilize P when grown in the culture medium or/and growth substances. Earlier studies demonstrated that soil microorganisms were involved in a series of processes that participated in the transformation and cycling of P. In particular, microorganisms are capable of efficiently converting insoluble P into accessible forms, thus directly releasing P ions (i.e., phosphate PO$_4^{3-}$) into the solution (Rodríguez and Fraga, 1999; Zhu, 2002). These findings are in agreement with our study, in which inoculation of A. aculeatus improved the soluble P content of medium and soil. It is universally accepted that the mechanism of phosphate solubilization by microorganisms is attributed mainly to the secretion of organic acids.
Table 2. The photosynthetic parameters of perennial ryegrass leaves that were subjected to phosphorus deficiency stress after inoculation with *Aspergillus aculeatus* (F) using a pulse-amplitude modulation fluorometer.

| Parameter                        | Treatment* | Data extracted from the recorded OJIP fluorescence transient curves | Specific energy fluxes (per active PSII reaction center) | Performance indexes [PI (combination of parameters)] |
|----------------------------------|------------|---------------------------------------------------------------------|--------------------------------------------------------|-----------------------------------------------------|
|                                  | CK         | F | Ca₃(PO₄)₂ | Ca₃(PO₄)₂ + F | Definition                                               |                                                   |
| Fluorescence at time t after onset (Fo) of actinic illumination | 0.29 a² | 0.29 a | 0.30 a | 0.29 a | Electron transport (ET₀) flux (further than Quinone A to quinone B) per RC |
| Fluorescence value at the J-step (Fj) of OJIP | 0.07 b | 0.80 ab | 0.81 ab | 0.82 a | Electron flux reducing (RE₀) end electron acceptors at the PSI acceptor side, per RC |
| Fluorescence value at the I-step (FI) of OJIP | 1.07 b | 1.08 b | 1.14 a | 1.14 a | Efficiency/probability with which a PSII trapped electron (ΨE₀) is transferred from quinoneA to quinoneB |
| Maximum (peak) fluorescence (Fm), when all photosystem II (PSII) reaction centers (RCs) are closed | 1.13 c | 1.15 c | 1.21 b | 1.23 a | Performance index for primary photochemistry (φP₀), namely F₀/Fₚ |
| Quantum yields and efficiencies/probabilities | 0.76 a | 0.76 a | 0.77 a | 0.77 a | Maximum quantum yield for primary photochemistry (φP₀), namely F₀/Fₚ |
| Quantum yield of the electron transport flux (φE₀) from quinoneA to quinoneB | 0.31 b | 0.34 ab | 0.31 b | 0.35 a | Quantum yield for reduction (φR₀) of end electron acceptors at the photosystem I (PSI) acceptor side |
| Quantum yield for reduction (φR₀) of end electron acceptors at the photosystem I (PSI) acceptor side | 0.07 a | 0.07 a | 0.06 a | 0.07 a | Quantum yield for reduction (φR₀) of end electron acceptors at the photosystem I (PSI) acceptor side |
| Efficiency/probability with which a PSII trapped electron (ΨE₀) is transferred from quinoneA to quinoneB | 0.40 b | 0.39 b | 0.40 b | 0.46 a | Efficiency/probability with which a PSII trapped electron (ΨE₀) is transferred from quinoneA to quinoneB |
| Specific energy fluxes (per active PSII reaction center) | 1.35 ab | 1.44 a | 1.29 b | 1.42 ab | Electron transport (ET₀) flux (further than Quinone A) per RC |
| Electron flux reducing (RE₀) end electron acceptors at the PSI acceptor side, per RC | 0.19 b | 0.22 b | 0.21 b | 0.30 a |
| Performance indexes [PI (combination of parameters)] | 0.46 b | 0.44 b | 0.46 b | 0.56 a | Performance index on absorption basis (PIₐ₅₈₅₅₈) |
| Performance index for energy conservation from excitation to the reduction of PSI end acceptors (PI_total) | 0.10 b | 0.10 b | 0.11 b | 0.15 a |

* Treatments consisted of 0 and 3 g Ca₃(PO₄)₂ added per pot. The chlorophyll a fluorescence transient was determined using a pulse-amplitude modulation fluorometer on the fourth fully expanded leaves after a 30-min dark adaptation. The curves were evaluated with a red light of 3000 μmol m⁻² s⁻¹ to acquire the maximum fluorescence intensity. The chlorophyll a fluorescence emission was induced by the strong light pulses and then was determined and digitized between 10 μs and 320 ms; CK = control [i.e., no Ca₃(PO₄)₂ and F in soil].

Different letters indicate significant differences under four different treatments based on Student-Newman-Keuls test combining one-way analysis of variance (P < 0.05).

(Dobbelaere et al., 2003; Pratibha and Arvind, 2009; Taktek et al., 2015). In addition, it was previously reported that organic acid played a central role in the process of solubilizing P (Pratibha and Arvind, 2009) by using their carboxyl and hydroxyl groups to chelate the cations (mainly Ca²⁺), and thereby releasing P in soluble form (Kpomblekou and Tabatabai, 1994). In our results, an evident increase in the contents of tartaric acid, citric acid, malic acid, and aminoacetic acid were observed in *A. aculeatus*–inoculated liquid medium, compared with nonfungi treatment condition. Particularly, we observed that the production of organic acids was accompanied by a drop of the pH in the liquid medium, which further reaffirmed the phosphate solubilization of *A. aculeatus*. Collectively, these observations and our results suggested that *A. aculeatus* was capable of contributing toward the conversion of the insoluble P (such as tricalcium phosphate) into available P.

To elucidate whether the *A. aculeatus* could enhance the performance of plants subjected to P-impoverished soils, we focused on the growth index, such as growth rate and turf quality. Phosphorus is an essential macronutrient necessary for plant growth, whose low availability or deficiency is a primary constraint to crop yield around the world (Imen et al., 2015; Wang et al., 2010). Previous studies have reported that beneficial effects of P on cell division and root elongation improved plant growth and yield (Sharma et al., 2012). In our results, we found that application of the *A. aculeatus* could promote the growth of plants in the absence of P source compared with control, which is supported by the elevated growth rate and turf quality. Nevertheless, the promoting effect of fungus was enhanced remarkably on the growth rate and turf quality in the presence of tricalcium phosphate. This phenomenon might be attributed to the P-solubilizing capability of *A. aculeatus*, which enhanced the absorption and utilization of P nutrition elements and thus facilitated the growth of plants. Our results accord with those of Imen et al. (2015), who suggested the decisive importance of inoculation of *Rhizobia* on improving plant growth and the P uptake. In addition, other previous reports also emphasized that P-solubilizing microorganisms, such as genera *Bradyrhizobium* and *Aspergillus*, could elevate the growth of radish (*Raphanus sativus*), maize (*Zea mays*), lettuce (*Lactuca sativa*), and wheat (*Triticum turgidum*) (Antoun et al., 1998; Chabot et al., 1996, 1998; Sharma et al., 2012). In the research of Chabot et al. (1996), it was demonstrated that phosphate solubilization effect of strains appears to be the most common mechanism of promoting plant growth. Overall, according to the observations and findings, we suggested that *A. aculeatus* as inoculant is involved in increasing plant P acquisition and growth.
Table 3. The parameters for analysis of amino, organic, and fatty acids, as well as sugars, and myo-inositol using gas chromatography–mass spectrometry of perennial ryegrass leaves that were subjected to phosphorus deficiency stress after inoculation with *Aspergillus aculeatus* (F).

| Metabolite          | CK       | F         | Ca3(PO4)2 | Ca3(PO4)2 + F |
|---------------------|----------|-----------|-----------|---------------|
|                     | (mean ± SD) |           |           |               |
| **Amino acids**     |          |           |           |               |
| Valine              | 0.0381 ± 0.000 b | 0.0473 ± 0.005 b | 0.0416 ± 0.001 b | 0.0565 ± 0.006 a |
| Isoleucine          | 0.0201 ± 0.001 b | 0.0252 ± 0.001 a | 0.0193 ± 0.001 b | 0.0240 ± 0.003 a |
| Serine              | 0.1660 ± 0.003 b | 0.1731 ± 0.007 b | 0.1765 ± 0.010 b | 0.2052 ± 0.016 a |
| Threonine           | 0.0926 ± 0.001 b | 0.1046 ± 0.007 b | 0.0816 ± 0.001 b | 0.1112 ± 0.010 a |
| Proline             | 0.1177 ± 0.016 d | 0.2352 ± 0.009 c | 0.4945 ± 0.038 b | 0.6630 ± 0.029 a |
| Trihydroxybutyric acid | 0.0190 ± 0.001 a | 0.0194 ± 0.002 a | 0.0075 ± 0.001 b | 0.0091 ± 0.002 b |
| Glutamic acid       | 0.0738 ± 0.003 a | 0.0694 ± 0.005 a | 0.0482 ± 0.017 b | 0.0563 ± 0.008 b |
| Tyrosine            | 0.0061 ± 0.001 a | 0.0075 ± 0.001 a | 0.0064 ± 0.001 a | 0.0094 ± 0.002 a |
| **Organic acids**   |          |           |           |               |
| Succinic acid       | 0.0249 ± 0.003 a | 0.0186 ± 0.003 b | 0.0188 ± 0.002 b | 0.0178 ± 0.002 b |
| Fumaric acid        | 0.0212 ± 0.001 b | 0.0282 ± 0.000 a | 0.0159 ± 0.001 c | 0.0130 ± 0.002 d |
| Malic acid          | 3.7538 ± 0.665 a | 4.0593 ± 0.372 a | 4.7866 ± 0.171 a | 3.9763 ± 0.322 a |
| Citric acid         | 1.0157 ± 0.024 a | 0.7201 ± 0.047 a | 0.7519 ± 0.047 b | 0.5840 ± 0.003 c |
| Ribonic acid        | 0.0189 ± 0.008 a | 0.0104 ± 0.001 a | 0.0095 ± 0.000 a | 0.0114 ± 0.001 a |
| Phthalic acid       | 0.0605 ± 0.014 a | 0.0304 ± 0.006 b | 0.0181 ± 0.004 b | 0.0189 ± 0.002 b |
| Glyceric acid       | 0.0374 ± 0.017 a | 0.0358 ± 0.005 a | 0.0442 ± 0.002 a | 0.0405 ± 0.000 a |
| Gluconic acid       | 0.0101 ± 0.001 a | 0.0092 ± 0.000 a | 0.0133 ± 0.003 a | UD |
| **Sugars**          |          |           |           |               |
| Sucrose             | 2.8234 ± 0.278 a | 3.0753 ± 0.447 a | 3.1705 ± 0.884 a | 2.0877 ± 0.113 a |
| Galactose           | 0.0875 ± 0.005 ab | 0.0921 ± 0.011 a | 0.0672 ± 0.008 b | 0.0729 ± 0.011 ab |
| Fructose            | 0.4159 ± 0.004 a | 0.4329 ± 0.135 a | 0.3112 ± 0.031 a | 0.1137 ± 0.014 b |
| Psicose             | 0.2552 ± 0.003 a | 0.1952 ± 0.108 a | 0.1798 ± 0.011 a | UD |
| Mannose             | 0.0879 ± 0.042 c | 0.7354 ± 0.046 bc | 0.1715 ± 0.047 b | 2.366 ± 0.064 a |
| Allose              | 0.2754 ± 0.095 a | 0.0446 ± 0.007 b | 0.0652 ± 0.025 b | 0.0416 ± 0.007 b |
| **Fatty acids**     |          |           |           |               |
| Hexadecanoic acid   | 0.1757 ± 0.032 a | 0.1007 ± 0.013 a | 0.1570 ± 0.049 a | 0.0934 ± 0.043 a |
| Octadecanoic acid   | 0.0688 ± 0.015 a | 0.0310 ± 0.013 b | 0.0281 ± 0.008 b | 0.0339 ± 0.020 b |
| Glycerol            | 1.4322 ± 0.103 a | 1.2380 ± 0.032 b | 1.1038 ± 0.060 c | 1.0881 ± 0.000 c |
| **Myo-inositol**    | 0.3345 ± 0.018 a | 0.3826 ± 0.041 a | 0.3586 ± 0.008 a | 0.3647 ± 0.021 a |

*4Treatments consisted of 0 and 3 g Ca3(PO4)2 added to per pot. The fully expanded perennial ryegrass leaves (±0.3 g) were extracted and determined by gas chromatography–mass spectrometry using the retention time coupled with available compound libraries. Relative quantification of the metabolite was calculated according to the internal standard (ribitol); CK = control [i.e., no Ca3(PO4)2 and F in soil].

*5Different letters indicate significant differences under four different treatments based on Student-Newman-Keuls test combining one-way analysis of variance (*P* < 0.05).

UD = undeterminable.

Plants that are exposed to P-impoverished soils would elicit the alteration in photosynthesis that is extremely sensitive to any environment changes (Lima et al., 2000; Sun et al., 2009). Several studies also highlighted the effects of P on photosynthesis, whereby there was a remarkable decline in the photosynthesis that is extremely sensitive to any environment changes (Lima et al., 2000; Sun et al., 2009). Chlorophyll *a* fluorescence is a useful tool that could probe the impact of abiotic stress on photosynthesis (Kalaji et al., 2011). Consistent with those findings, our results showed that chlorophyll *a* fluorescence transient (OJIP curve) of plant leaves significantly reduced in the control [i.e., no Ca3(PO4)2 and *A. aculeatus* in medium], compared with Ca3(PO4)2 treatment or Ca3(PO4)2 + *A. aculeatus* treatment. This result might be attributed to the inadequate supply of P, which caused a decrease in leaf area, altered leaf chlorophyll, and protein content (Li et al., 2006; Plesnicar et al., 1994; Usuda, 1995). It is noteworthy that the OJIP had an obvious increase in Ca3(PO4)2 + *A. aculeatus* group, compared with Ca3(PO4)2 treatment, which suggests the important role of *A. aculeatus* in dissolving P and hence contributing positively toward photosynthesis. Furthermore, parallel to enhanced OJIP, photosynthetic parameters, such as φE0, ΨE0, RE0/RC, PIABS, and PI tot, showed an evident increase in Ca3(PO4)2 + *A. aculeatus* group, compared with the *A. aculeatus* group or Ca3(PO4)2 group. This result indicated that increased availability of the P content induced by *A. aculeatus* was closely associated with enhanced photosynthetic efficiency accompanied by the elevated photosynthetic parameters and OJIP curve. Consistent with those observations, a study by Yu et al. (2010) found that the net photosynthetic rate of leaves was enhanced significantly by phosphate-solubilizing bacteria (Stevenson, 1994; Yu et al., 2010). Therefore, those results implied that *A. aculeatus* through its phosphate solubilization regulated photosynthesis of plant under P-impoverished soils.

A number of studies have illustrated that metabolites are considered as an important tool in exploring plant stress.
responses, including shortage in nutrients, ion stresses, oxidative stress, and so on (Guy et al., 2008; Obata and Fernie, 2012; Shulaev et al., 2008). A previous study showed that environmental changes could disrupt and alter the metabolites of plants (Widodo et al., 2009). It was confirmed that leaf amino acid content was significantly reduced or enhanced (Julia et al., 2015; Lima et al., 2000), organic acid content increased (Julia et al., 2015), and sugar content also increased (Ganie et al., 2015; Hernandez and Vance, 2007) under P-deficiency condition. Consistent with those reports, our results indicated that leaf amino acid content increased or decreased dramatically and organic acid content increased distinctly when the plants were subjected to P-deficiency stress. Most of the detected amino acids were remarkably decreased in P-deficient levels, which are mainly due to the inadequacy of energy required for the synthesis of amino acids in P-deficiency plants, thereby contributing to the drop of amino acid content (Lima et al., 2000). Conversely, an accumulation of only a few of the amino acids might be because of accelerated protein degradation and suppressed protein synthesis under P-limiting conditions (Ganie et al., 2015). Moreover, it was evidenced that organic acids acted as crucial roles in the plant adaptation to nutrient deficiency (Wang and Shen, 2006), which explains the enhancement of organic acids in plant leaves exposed to P-deficiency condition. In addition, another study confirmed that those organic acids were preferred reserves for P-deficiency tolerance (Li et al., 2007). Interestingly, in the Ca3(PO4)2 + F group, amino acid and organic acid content exhibited a converse trend when compared with other treatments (i.e., P-deficiency condition), which might be the result of phosphate solubilization of A. aculeatus that enhanced the acquisition and utilization efficiency of P under P-deficiency stress. Therefore, the metabolite profiles of shoots were used to gather insight of the metabolic adjustments to plants faced with the P-deficient environment, and A. aculeatus may be involved in modulating metabolite accumulation.

In conclusion, our results demonstrated that A. aculeatus has the capacity to solubilize insoluble P into soluble P and accelerate plant growth under low P stress. According to our findings, we put forward the mechanism of A. aculeatus to solubilize insoluble P in a liquid medium mainly due to organic acids secreted by fungi. Furthermore, A. aculeatus promotes the growth of perennial ryegrass subjected to P-impoverished soils and the main performance is in the following three aspects: 1) A. aculeatus could promote plant growth rate and turf quality; 2) A. aculeatus can enhance plant photosynthetic efficiency; and 3) A. aculeatus could regulate synthesis and accumulation of metabolites (amino acids, organic acids, and sugars). We highlight the high potential of this fungus in solubilizing insoluble P and appraise the potential of the fungus to be used in commercial biofertilizers.

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