Microstructure Analysis and Mechanical Properties of Low Alloysed Steel with Retained Austenite Obtained by Heat Treatment

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Four different heating temperatures in the range of 770 °C – 950 °C were used for laboratory heat treatment of low carbon low alloyed steel. Chemical composition of the steel was based on the most common TRIP steel concept, only the silicon content was lowered to 0.6% and it was partially replaced by 1.4% of aluminium. The steel was further micro-alloyed by niobium. Two different ways of cooling were applied to the samples. The first set was cooled to 425 °C in a salt bath with the temperature of 200 °C, the second set was cooled to 425 °C in a salt bath heated to the temperature of 400 °C. In this way, two distinctive cooling rates were achieved for every soaking temperature. Once the samples reached 4250 °C, they were in all cases removed to the furnace for 20 minute hold at the temperature of 425 °C. The final cooling was carried out in air. Resulting microstructures were analysed by scanning electron microscopy and consisted of various amounts of ferrite, bainite and retained austenite. Tensile strength in the range of 750 – 908 MPa was obtained with total elongation of 33-42%.

Keywords: TRIP steel, Aluminium, Heat treatment, Retained austenite

1 Introduction

TRIP (transformation induced plasticity) steels are high strength low alloyed steels that can be produced either by heat or thermo-mechanical treatment [1-3]. Heat treatment typically consists of two step annealing with the second hold in the temperature region of bainitic transformation [4-6]. The aim of the processing is to prepare multiphase microstructure containing ferrite, carbide-free bainite and retained austenite [7-10]. This microstructure has the potential to utilize TRIP effect, when metastable retained austenite transforms to martensite during plastic deformation. When gradual transformation occurs during straining, the offset of necking is postponed and uniform plastic deformation of the steel is enhanced and ultimate tensile strength is increased at the same time. The presence of hard martensitic islands and thin films in the microstructure results in further increase of the strength of the steel. Due to this effect, TRIP steels can be used to decrease the weight of formed parts in automotive industry.

First generation of TRIP steels was based on 0.2-0.4% C- 1 – 2% Mn – 1 – 2% Si alloying concept, which ensured stabilization of around 10-15% of retained austenite and interesting combinations of tensile strength and ductility [1,11]. Manganese was used to support retained austenite stabilization and silicon was added mainly to suppress carbide precipitation during the hold in bainite transformation temperature region. Retention of austenite is achieved by combined effects of geometrical and chemical stabilization and carbon content in retained austenite is therefore typically high (around 1 – 2 %) [1, 7-10]. This is the reason why carbide formation is undesirable during the processing, as it consumes carbon needed for stabilization of retained austenite. Higher carbon content means higher stability of a particular island or lath of retained austenite, but does not necessarily result also in better mechanical properties. Particularly small islands and thin films of retained austenite tend to possess higher carbon contents and might be therefore over-stabilized. It means that they do not transform at all during straining and thus do not contribute to the enhancement of mechanical properties. Higher silicon contents used in TRIP steels presented problems with welding and galvanizing, which are both important for an automotive industry. Other alloying concepts have been investigated to improve the performance of the steel in these two areas without loss of mechanical properties [12-17]. One of the solutions seemed to be partial substitution of silicon by aluminium. Complete substitution was not successful, aluminium has similar effect on austenite stability and solid solution strengthening as silicon, however the effect is weaker and the steel possessed lower strength [7-9].

2 Experimental Program

Tab. 1 Chemical composition of experimental steel in weight %

| C | Mn | Si | P | S | Al | Nb |
|---|----|----|---|---|----|----|
| 0.2 | 1.5 | 0.6 | 0.008 | 0.005 | 1.3 | 0.06 |

Fig. 1 TTT diagram of experimental steel calculated in JMatPro software
Low alloyed experimental steel (Tab. 1) was prepared as a 250 kg ingot using a laboratory vacuum furnace. The ingot was divided into four parts and re-forged at the temperature of 1150 °C. Forged bars were subsequently annealed for two hours at 950 °C and air cooled to room temperature.

Cylindrical samples with the length of 5 cm were cut directly from the bars and used for heat treatment in the laboratory furnace. Soaking temperatures of 770 °C, 800 °C, 850 °C and 900 °C were applied to the steel and soaking hold was always 20 minutes. The samples were then cooled in salt bath heated to the temperature of either 200 °C or 400 °C to obtain two different cooling rates. Cooling rate of the samples cooled in salt bath with the temperature of 200 °C was around 27 °C/s, while samples cooled in 400 °C salt bath reached lower cooling rates around 13 °C/s. According to previous works carried out at the steel with similar chemical composition without aluminium [3, 13, 14] and TTT (time temperature transformation) diagram of experimental steel calculated in JMAtPro (Fig. 1), even the lower cooling rate should be high enough to prevent pearlite formation during the cooling. The temperature of every sample was monitored during heat treatment by a set of thermocouples of MTC 130/12 a DKO 4 furnaces were used for heat treatment of the steel.

Metallographic cross sections were prepared from heat treated samples in transverse direction (perpendicular to the axes of the bars) and etched in 3% Nital. The resulting microstructures were analysed using a Zeiss EVO 25 scanning electron microscope with a LaB6 cathode and an Olympus BX61 light microscope. The volume fraction of the retained austenite was determined by X-ray diffraction phase analysis using an AXS Bruker D8 Discover automatic powder diffractometer with a HiSTAR detector and Co lamp using integrated intensities of (200) ferrite peak and (111), (002) and (022) austenite peaks. Mechanical properties were evaluated by tensile test of flat dog-boned sub-sized samples with gauge length of 5 mm and the cross section of 2 x 1.2 mm. Carbon content in retained austenite was calculated considering the effect of alloying elements according to [13]:

\[ a = 3.572 + 0.0012 \text{ Mn} - 0.00157 \text{ Si} + 0.0056 \text{ Al} + 0.033 \text{ C} \]  

Where: \( a \) = lattice parameter of the retained austenite calculated from the three austenite peaks measured by X-ray diffraction phase analysis. The amounts of alloying elements are given in weight percent.

3 Results and discussion

3.1 Samples cooled in salt bath with 200 °C

All three samples processed with cooling in salt bath with the temperature of 200 °C possessed bainite-based microstructures with 12-17% of retained austenite with yield strengths of 568-658 MPa, good tensile strengths in the region of 814 – 843 MPa and high total elongation of 40-42% (Tab. 2). Carbon contents in the retained austenite were also high, reaching the values between 1.892% and 2.052%, suggesting high chemical stabilization of the retained austenite.

| Soaking temperature [°C] | Salt bath temperature [°C] | Re [MPa] | Rm [MPa] | A [%] | RA [%] | C in RA [%] |
|-------------------------|---------------------------|----------|----------|-------|--------|-----------|
| 850                     | 200                       | 568      | 830      | 42    | 15     | 1.895     |
| 900                     |                            | 591      | 814      | 40    | 17     | 1.892     |
| 950                     |                            | 658      | 843      | 40    | 12     | 2.052     |
| 770                     | 400                       | 396      | 750      | 41    | 18     | 1.855     |
| 850                     |                            | 410      | 815      | 39    | 16     | 1.762     |
| 900                     |                            | 435      | 856      | 37    | 11     | 1.601     |

Heat treatment with soaking temperature of 850 °C produced bainitic-ferritic microstructure with small amount of proeutectoid ferrite and 15% of retained austenite. Bainitic microstructure consisted of the mixture of lath and granular bainite, both of them being carbide-free. This microstructure reached the highest total elongation of 42% combined with relatively high tensile strength of 830 MPa.

Increasing soaking temperature from 850 °C to 900 °C decreased the amount of proeutectoid ferrite formed at prior austenite grain boundaries (Fig. 2 - Fig. 4). The resulting microstructure was predominantly bainitic with 17% of retained austenite. Carbon content in the retained austenite was the same as in the microstructure soaked at 850 °C. Larger areas of bainitic ferrite were observed in the microstructure, which obtained lower tensile strength of 814 MPa and total elongation of 40%.

Further increase of soaking temperature to 950 °C decreased even the thickness of bainitic ferrite laths (Fig. 2, Fig. 5). Lath morphology of the bainite was prevailing in this microstructure. The lowest volume fraction of retained austenite (12%) was measured in the sample soaked at the highest temperature of 950 °C (Tab. 2), however it contained the highest amount of carbon above 2%. This microstructure achieved higher yield and tensile strengths of 658 MPa and 843 MPa respectively, than the previous two samples processed at lower temperature.
3.2 Samples cooled in salt bath with 400 °C

The lowest soaking temperature of 770 °C was in the intercritical ferrite-austenite region, as can be seen from the comparison of light micrographs of this sample with the one soaked at 850 °C (Fig. 6 – Fig. 8). There are large grains of original polygonal ferrite in the sample soaked at 770 °C, which did not transform to austenite during soaking. SEM details from a bainitic region show the mixture of granular and lath bainite (Fig. 7). There are also small bulky islands of retained austenite or M-A constituent, particularly at the edges of bainitic areas. Larger amount of polygonal ferrite in the final microstructure was responsible for the drop of yield and ultimate tensile strengths to 396 MPa and 750 MPa, respectively. Due to the small carbon solubility in ferrite, more carbon remained for retained austenite stabilization, increasing its volume fraction to 18%. Carbon content in retained austenite was also relatively high, reaching 1.855%.

There was practically no polygonal ferrite in the microstructure obtained by soaking at 850 °C and slow cooling (Fig. 8). The microstructure was predominantly bainitic and consisted of the mixture of various bainitic morphologies, ranging from larger packets of lath bainite to granular bainite and 16% of retained austenite. An average combination of tensile strength of 815 MPa and 39% total elongation was obtained for this microstructure. Carbon content in retained austenite decreased significantly to 1.762% in comparison to the previous treatment with lower soaking temperature.

The soaking temperature of 900 °C combined with lower cooling rate resulted in the formation of large islands of M-A constituent with complex shapes and distinctive areas with martensitic laths (Fig. 6, Fig. 9). The amount of retained austenite found in this microstructure was only 11%, which was the lowest value obtained in this work. In combination with the lowest carbon content of 1.6%, it suggest the lowest stability of retained austenite in this sample. High amount of martensite would contribute to the highest strength and lowest total elongation of the sample. There are some fine grains of proeutectoid ferrite, which were not observed in the previous sample; however the amount of bainitic ferrite seems to be lower than in the case of sample soaked at 850 °C.
Retained austenite volume fraction decreased with increasing soaking temperature for the samples cooled in salt bath heated to 400 °C. The correlation of decreasing retained austenite volume fraction to decreasing its carbon content was observed. These results are in agreement with results obtained by Wang et al. [12] at 0.25C-1.23Si-2.09Mn-2.92Al steel. On the other hand, no similar correlation between soaking temperature, retained austenite volume fraction and retained austenite carbon content was found in the samples cooled in salt bath heated to 200 °C, where the highest carbon content above 2% was determined for the lowest volume fraction of retained austenite of 12%.

4 Conclusions

The volume fraction of retained austenite was for all treatments between 11 and 18 %. With the exception of sample soaked at 770 °C, tensile strengths of 815-850 MPa were obtained. Increasing temperature of soaking hold decreased the amount of proeutectoid ferrite at prior austenite grain boundaries. Bainite-based microstructures with retained austenite carbon contents were obtained by the treatments with cooling at 27 °C/s carried out in 200 °C salt bath. There was more ferrite and ferritic islands and laths were thicker and longer in the samples cooled at 13 °C/s in 400 °C salt bath. The microstructures obtained by slower cooling were generally coarser, retained austenite laths were also longer, retained austenite carbon contents were lower and lath bainite morphology was more pronounced. Sample processed with soaking temperature of 900 °C and slower cooling had distinctively larger islands of M-A constituent.

For the samples cooled in the 400 °C salt bath, there was a distinctive trend in mechanical properties and retained austenite volume fraction, which was not seen in the first set of quicker cooled samples. Yield and ultimate tensile strengths both increased with increasing soaking temperature, while retained austenite volume fraction and total elongation decreased at the same time. Retained austenite carbon content decreased with decreasing retained austenite volume fraction.

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