Approach to thermal equilibrium in harmonic crystals with polyatomic lattice

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Abstract

We study transient thermal processes in infinite harmonic crystals with complex (polyatomic) lattice. Initially particles have zero displacements and random velocities such that distribution of temperature is spatially uniform. Initial kinetic and potential energies are different and therefore the system is far from thermal equilibrium. Time evolution of kinetic temperatures, corresponding to different degrees of freedom of the unit cell, is investigated. It is shown that the temperatures oscillate in time and tend to generally different equilibrium values. The oscillations are caused by two physical processes: equilibration of kinetic and potential energies and redistribution of temperature among degrees of freedom of the unit cell. An exact formula describing these oscillations is obtained. At large times, a crystal approaches thermal equilibrium, i.e. a state in which the temperatures are constant in time. A relation between equilibrium values of the temperatures and initial conditions is derived. This relation is refereed to as the non-equipartition theorem. For illustration, transient thermal processes in a diatomic chain and graphene lattice are considered. Analytical results are supported by numerical solution of lattice dynamics equations.

Keywords: thermal equilibrium; stationary state; approach to equilibrium; polyatomic lattice; complex lattice; kinetic temperature; harmonic crystal; transient processes; equipartition theorem; non-equipartition theorem; temperature matrix.

1 Introduction

In classical systems at thermal equilibrium, the kinetic energy of thermal motion of atoms is usually equally shared among degrees of freedom. This fact follows from the equipartition theorem [26, 62]. The theorem allows to characterize thermal state of the system by a single scalar parameter, notably the kinetic temperature, proportional to kinetic energy of thermal motion.

Far from thermal equilibrium, kinetic energies, corresponding to different degrees of freedom, can be different [8, 17, 24, 25, 27, 45]. Therefore in many works several temperatures are introduced [8, 17, 30, 31, 45]. For example, it is well known that temperatures of a lattice and electrons in solids under laser excitation are different (see e.g. a review paper [45]). Two temperatures are also observed in molecular dynamics simulations of shock

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waves. In papers [25, 24, 27, 60] it is shown that kinetic temperatures, corresponding to thermal motion of atoms along and across the shock wave front are different. Different nonequilibrium temperatures of sublattices of methylammonium lead halide are reported in papers [10, 17]. In papers [32, 33], stationary heat transfer in a harmonic diatomic chain connecting two thermal reservoirs is considered. It is shown that temperatures of sublattices at the nonequilibrium steady state are different.

In the absence of external excitations, the nonequilibrium system tends to thermal equilibrium. Approach to thermal equilibrium is accompanied by several physical processes. Distribution of velocities tends to Gaussian [23, 14, 35, 43, 57]. The total energy is redistributed among kinetic and potential forms [1, 35, 37, 41]. Kinetic energy is redistributed between degrees of freedom [41]. The energy is redistributed between normal modes [52]. These processes, except for the last one, are present in both harmonic and anharmonic systems [1, 14, 35, 41, 43, 57]. In harmonic crystals, energies of normal modes do not equilibrate. However distribution of kinetic temperature in infinite harmonic crystals tends to become spatially and temporary uniform [23, 57, 42]. Therefore the notion of thermal equilibrium is widely applied to infinite harmonic crystals [7, 13, 14, 15, 22, 29, 43, 57, 59].

Approach to thermal equilibrium in harmonic crystals is studied in many works [7, 13, 14, 15, 22, 23, 29, 35, 37, 40, 41, 42, 43, 44, 50, 57]. Various aspect of this process are studied, including existence of the equilibrium state [43], ergodicity [59, 22], convergence of the velocity distribution function [7, 35, 14, 15], evolution of entropy [28, 29, 56] etc. In the present paper, we focus on the behavior of the main observable, notably kinetic temperature (temperatures).

Two different approaches for analytical treatment of harmonic crystals are presented in literature. One approach employs an exact solution of equations of motion [18, 23, 35, 29, 44]. Given known the exact solution, kinetic temperature is calculated as mathematical expectation of corresponding kinetic energy. For example, in pioneering work of Klein and Prigogine [35], transition to thermal equilibrium in an infinite harmonic one-dimensional chain with random initial conditions is investigated. Using exact solution derived by Schrödinger [55], it is shown that kinetic and potential energies of the chain oscillate in time and tend to equal equilibrium values [35]. Another approach uses covariances\(^1\) of particle velocities and displacements as main variables. For harmonic crystals, closed system of equations for covariances can be derived in steady [32, 47, 53] and unsteady cases [37, 40, 41, 42, 19, 48]. Solution of these equations describes, in particular, time evolution of kinetic temperature. In papers [37, 40, 41, 42], this idea is employed for description of approach to thermal equilibrium in harmonic crystals with simple (monoatomic) lattice\(^2\). In particular, monoatomic one-dimensional chains [2, 37] and two-dimensional lattices [42, 40, 41] have been covered.

In the present paper, we study approach towards thermal equilibrium in an infinite harmonic crystal with polyatomic lattice\(^3\). Our main goals are to describe time evolution

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\(^1\)Covariance of two centered random values is equal to mathematical expectation of their product.

\(^2\)A lattice is referred to as simple lattice, if it coincides with itself under shift by a vector connecting any two particles.

\(^3\)Polyatomic lattice consists of several simple monoatomic sublattices. For example, graphene lattice consists of two triangular sublattices.
of kinetic temperatures, corresponding to different degrees of freedom of the unit cell, and to calculate equilibrium values of these temperatures.

The paper is organized as follows. In section 2, equations of motion for the unit cell are represented in a matrix form. It allows to cover monoatomic and polyatomic lattices with interaction of an arbitrary number of neighbors and harmonic on-site potential. In section 3, approach to thermal equilibrium is considered. An equation describing the behavior of kinetic temperatures, corresponding to different degrees of freedom of the unit cell, is derived. An exact solution of this equation is obtained. In section 4, an expression relating equilibrium values of the temperatures with initial conditions is derived. In sections 5, 6, approach to thermal equilibrium in a diatomic chain and graphene lattice are studied. Obtained results are exact in the case of spatially uniform distribution of temperature in harmonic crystals. Implications of the nonuniform temperature distribution and anharmonic effects are discussed in the last section.

2 Equations of motion and initial conditions

We consider infinite crystals with complex (polyatomic) lattice in \(d\)-dimensional space, \(d = 1, 2, 3\). In this section, equations of motion of the unit cell are written in a matrix form, convenient for analytical derivations.

Unit cells of the lattice are identified by position vectors, \(x\), of their centers\(^4\). Each elementary cell has \(N\) degrees of freedom \(u_i(x), i = 1, \ldots, N\), corresponding to components of particle displacements. The components of displacements form a column:

\[ u(x) = (u_1, u_2, \ldots, u_N)^\top, \quad (1) \]

where \(\top\) stands for the transpose sign.

Particles from the cell \(x\) interact with each other and with particles from neighboring unit cells, numbered by index \(\alpha\). Vector connecting the cell \(x\) with neighboring cell number \(\alpha\) is denoted \(a_\alpha\). Centers of unit cells always form a simple lattice, therefore numbering can be carried out so that vectors \(a_\alpha\) satisfy the identity:

\[ a_\alpha = -a_{-\alpha}. \quad (2) \]

Here \(a_0 = 0\). Vectors \(a_\alpha\) for a sample lattice are shown in figure 1.

Consider equations of motion of the unit cell. In harmonic crystals, the total force acting on each particle is represented as a linear combination of displacements of all other particles. Using this fact, we write equations of motion in the form\(^5\):

\[ M\ddot{v}(x) = \sum_{\alpha} C_\alpha u(x + a_\alpha), \quad C_\alpha = C_{-\alpha}^\top, \quad (3) \]

where \(v = \dot{u}; u(x + a_\alpha)\) is a column of displacements of particles from unit cell \(\alpha\); \(M\) is diagonal \(N \times N\) matrix composed of particles’ masses; for \(\alpha \neq 0\) coefficients of \(N \times N\)

\(^4\)For analytical derivations, position vectors are more convenient than indices, because number of indices depends on space dimensionality.

\(^5\)Similar form of equations of motion is used in paper [50].
matrix $C_\alpha$ determine stiffnesses of springs connecting unit cell $x$ with neighboring cell number $\alpha$; matrix $C_0$ describes interactions of particles inside\(^6\) the unit cell $x$. Summation is carried out with respect to all unit cells $\alpha$, interacting with unit cell $x$ (including $\alpha = 0$).

Formula (3) describes motion of monoatomic and polyatomic lattices in one-, two-, and three-dimensional cases. For example, one-dimensional diatomic chain and two-dimensional graphene lattice are considered in sections 5, 6. Matrices $M, C_\alpha$ for these lattices are given by formulas (37), (49).

**Remark.** For $N = 1$ (one degree of freedom per unit cell), equation (3) governs dynamics of the so-called scalar lattices\(^7\), considered, for example, in papers [20, 42, 50, 51].

The following initial conditions, typical for molecular dynamics modeling [1], are considered:

$$u(x) = 0, \quad v(x) = v_0(x),$$

where $v_0(x)$ is a column of random initial velocities of particles from unit cell $x$. Components of $v_0(x)$ are random numbers with zero mean\(^8\) and generally different variances. The variances are independent of $x$. Initial velocities of particles in different unit cells are statistically independent, i.e. their covariance is equal to zero. Under these initial conditions, spatial distribution of statistical characteristics, e.g. kinetic temperature, is uniform.

Equations of motion (3) with initial conditions (4) completely determine dynamics of a crystal at any moment in time. The equations can be solved analytically using, for example, discrete Fourier transform. Resulting random velocities of the particles can be used for calculation of statistical characteristics, such as kinetic temperature. However in the following sections, we use another approach, which allows to formulate and solve

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\(^6\)Additionally, matrix $C_0$ can include stiffnesses of harmonic on-site potential.

\(^7\)In scalar lattices each particle has only one degree of freedom. This model is applicable to monoatomic one-dimensional chains with interactions of arbitrary number of neighbors and to out-of-plane motions of monoatomic two-dimensional lattices.

\(^8\)In this case mathematical expectations of all velocities are equal to zero at any moment in time.
equations for statistical characteristics of the crystal with deterministic initial conditions.

3 Approach to thermal equilibrium

Initial conditions (4) are such that initially kinetic and potential energies of the crystal are different (potential energy is equal to zero). Motion of particles leads to redistribution of energy among kinetic and potential forms. Therefore kinetic temperature, proportional to kinetic energy, changes in time. In this section, we derive a formula, exactly describing time evolution of kinetic temperatures, corresponding to different degrees of freedom of the unit cell. The formula shows that a crystal evolves towards a state in which the temperatures are constant in time. This state is further referred to as the thermal equilibrium.

3.1 Generalized kinetic energies. Kinetic temperature

In this section, we derive an equation, exactly describing time evolution of the kinetic temperatures during approach to thermal equilibrium.

We consider an infinite set of realizations of the same system. The realizations differ only by random initial conditions (4). This approach allows to introduce statistical characteristics such as kinetic temperatures.

In general, each degree of freedom of the unit cell has its own kinetic energy and kinetic temperature. Then in order to characterize thermal state of the unit cell, we introduce $N \times N$ matrix, $T$, further referred to as the temperature matrix:

$$k_B T_{ij} = \sqrt{M_i M_j \langle v_i v_j \rangle}, \quad (5)$$

where $M_i$ is $i$-th element of matrix $M$, equal to a mass corresponding to $i$-th degree of freedom; $k_B$ is the Boltzmann constant; brackets $\langle .. \rangle$ stand for mathematical expectation. Diagonal element, $T_{ii}$, of the temperature matrix is equal to kinetic temperature, corresponding to $i$-th degrees of freedom of the unit cell, i.e. $k_B T_{ii} = M_i \langle v_i^2 \rangle$. Off-diagonal elements characterize correlation between velocities, corresponding to different degrees of freedom.

We also introduce the kinetic temperature, $T$, proportional to the total kinetic energy of the unit cell:

$$T = \frac{1}{N} \text{tr} T = \frac{1}{N} \sum_{i=1}^{N} T_{ii}, \quad (6)$$

where $N$ is a number of degrees of freedom per unit cell; $\text{tr}(..)$ stands for trace of a matrix. In the case of energy equipartition, kinetic temperatures, corresponding to all degrees of freedom of the unit cell, are equal to $T$.

Neither kinetic temperature (6) nor temperature matrix (5) is sufficient for derivation of closed system of equations. Therefore we introduce generalized kinetic energy $K(x,y)$.

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9In computer simulations, mathematical expectation can be approximated by average over realizations with different random initial conditions.

10Trace of a square matrix is defined as a sum of diagonal elements.
defined for any pair of unit cells \( x \) and \( y \) as

\[
\mathbf{K}(x, y) = \frac{1}{2} \mathbf{M}^{\frac{1}{2}} \left( \mathbf{v}(x) \mathbf{v}(y)^\top \right) \mathbf{M}^{\frac{1}{2}}. \tag{7}
\]

Diagonal elements of matrix \( \mathbf{K}(x, x) \) are equal to mathematical expectations of kinetic energies, corresponding to different degrees of freedom of the unit cell, i.e. \( K_{ii}(x, x) = \frac{1}{2} M_i \left( v_i(x)^2 \right) \). The generalized kinetic energy is related to the temperature matrix (5) as

\[
\frac{1}{2} k_B T = \mathbf{K}(x, x). \tag{8}
\]

**Remark.** Notion of generalized kinetic energy for monoatomic lattices is introduced in papers [37, 40, 41, 42]. In these papers, the energy is defined for pairs of particles rather than pairs of unit cells.

We consider initial conditions (4) such that spatial distribution of all statistical characteristic is uniform. In this case the following identity is satisfied:

\[
\mathbf{K}(x, y) = \mathbf{K}(x - y). \tag{9}
\]

Argument \( x - y \) is omitted below for brevity. In Appendix I, it is shown that the generalized kinetic energy, \( \mathbf{K}(x - y) \), satisfies equation:

\[
\dddot{\mathbf{K}} - 2 \left( \mathbf{L} \dddot{\mathbf{K}} + \dddot{\mathbf{K}} \mathbf{L} \right) + \mathbf{L}^{2} \mathbf{K} - 2 \mathbf{L} \mathbf{K} \mathbf{L} + \mathbf{K} \mathbf{L}^{2} = 0, \tag{10}
\]

Here \( \mathbf{L}^{2} \mathbf{K} = \mathbf{L} (\mathbf{L} \mathbf{K}) \). Formula (10) is equivalent to an infinite system of ordinary differential equations. It exactly describes the evolution of generalized kinetic energy in any harmonic lattice.

**Remark.** A particular case of equation (10) for a one-dimensional monoatomic harmonic chain with nearest-neighbor interactions was originally derived in paper [36].

**Remark.** Consider physical meaning of difference operator \( \mathbf{L} \). Using this operator, equation of motion (3) is represented as

\[
\mathbf{M}^{\frac{1}{2}} \dddot{\mathbf{u}}(x) = \mathbf{M}^{\frac{1}{2}} \sum_{\alpha} \mathbf{C}_{\alpha} \mathbf{u}(x + a_{\alpha}) = \mathbf{L} \left( \mathbf{M}^{\frac{1}{2}} \mathbf{u}(x) \right). \tag{11}
\]

Therefore \( \mathbf{L} \) is equal to operator in the right-hand side of equations of motion provided that the equations are written for \( \mathbf{M}^{\frac{1}{2}} \mathbf{u}(x) \).

Initial conditions for \( \mathbf{K} \), corresponding to initial conditions for particle velocities (4), have form:

\[
\mathbf{K} = \frac{1}{2} k_B T_0 \delta_D(x - y), \quad \dot{\mathbf{K}} = 0, \quad \dddot{\mathbf{K}} = \mathbf{L} \mathbf{K} + \mathbf{K} \mathbf{L}, \quad \dddot{\mathbf{K}} = 0, \tag{12}
\]

where \( \delta_D(0) = 1; \delta_D(x - y) = 0 \) for \( x \neq y; \ T_0 \) is the initial value of temperature matrix (5), independent of \( x \). Here the expression for \( \mathbf{K} \) follows from formula (8) and independence of initial velocities of different unit cells. Values \( \dddot{\mathbf{K}} \) and \( \dddot{\mathbf{K}} \) are proportional.
to covariance of displacements and velocities. Since initial displacements are equal to zero, then initial values of \( \dot{\mathbf{K}}, \ddot{\mathbf{K}} \) vanish. The expression for \( \ddot{\mathbf{K}} \) follows from formula (56), derived in Appendix I.

Thus the generalized kinetic energy \( \mathbf{K}(x - y) \) satisfies equation (10) with *deterministic* initial conditions (12). Given known the solution of this initial value problem, the temperature matrix, \( \mathbf{T} \), is calculated using formula (5).

### 3.2 Time evolution of the temperature matrix

In this section, we solve equation (10) for the generalized kinetic energy with initial conditions (12). The solution yields an *exact* expression for the temperature matrix at any moment in time.

The solution is obtained using the discrete Fourier transform with respect to variable \( x - y \). Vectors \( x - y \) form the same lattice as vectors \( x \). Then the vectors are represented as

\[
x - y = \sum_{j=1}^{d} a_j z_j \mathbf{e}_j,
\]

where \( a_j \mathbf{e}_j, j = 1, \ldots, d \) are basis vectors of the lattice; \( |\mathbf{e}_j| = 1 \); \( z_j \) are integers; \( d \) is space dimensionality. Direct and inverse discrete Fourier transforms for an infinite lattice are defined as

\[
\hat{\mathbf{K}}(\mathbf{k}) = \frac{1}{(2\pi)^d} \int_{0}^{2\pi} \cdots \int_{0}^{2\pi} \mathbf{K}(x - y) e^{-i\mathbf{k} \cdot (x - y)} d\mathbf{k}, \\
\mathbf{K}(x - y) = \int_{\mathbf{k}} \hat{\mathbf{K}}(\mathbf{k}) e^{i\mathbf{k} \cdot (x - y)} d\mathbf{k}.
\]

Here \( \hat{\mathbf{K}} \) is Fourier image of \( \mathbf{K} \); \( i^2 = -1 \); \( \mathbf{k} \) is wave vector; \( \mathbf{e}_j \) are vectors of the reciprocal basis, i.e. \( \mathbf{e}_j \cdot \mathbf{e}_k = \delta_{jk} \), where \( \delta_{jk} \) is the Kroneker delta; for brevity, the following notation is used:

\[
\int_{\mathbf{k}} \ldots d\mathbf{k} = \frac{1}{(2\pi)^d} \int_{0}^{2\pi} \cdots \int_{0}^{2\pi} dp_1 \ldots dp_d.
\]

Applying the discrete Fourier transform (14) in formulas (10), (12), yields equation

\[
\dddot{\mathbf{K}} + 2 \left( \Omega \ddot{\mathbf{K}} + \dot{\mathbf{K}} \Omega \right) + \Omega^2 \mathbf{K} - 2 \Omega \dot{\mathbf{K}} \Omega + \dot{\mathbf{K}} \Omega^2 = 0,
\]

\[
\Omega(\mathbf{k}) = - \sum_{\alpha} M^{-\frac{1}{2}} \mathbf{C}_{\alpha} M^{-\frac{1}{2}} e^{i\mathbf{k} \cdot \mathbf{a}_\alpha},
\]

with initial conditions\(^{11}\)

\[
\dot{\mathbf{K}} = \frac{1}{2} k_B \mathbf{T}_0, \quad \ddot{\mathbf{K}} = 0, \quad \dddot{\mathbf{K}} = - \Omega \dot{\mathbf{K}} - \dot{\mathbf{K}} \Omega, \quad \dddot{\mathbf{K}} = 0.
\]

Matrix \( \Omega \) in formula (16) coincides with the *dynamical matrix* of the lattice, derived in Appendix II (see formula (64)). Examples of matrix \( \Omega \) for two particular lattices are given by formulas (39), (51).

\(^{11}\)Here identities \( \Phi(\mathbf{K}(x - y + \mathbf{a}_\alpha)) = \hat{\mathbf{K}} e^{i\mathbf{k} \cdot \mathbf{a}_\alpha}, \Phi(\delta_D(x - y)) = 1, \Phi(\mathcal{L} \mathbf{K}) = - \Omega \hat{\mathbf{K}}, \Phi(\mathcal{L}^2 \mathbf{K}) = - \Omega \Phi(\mathcal{L} \mathbf{K}) = \Omega^2 \hat{\mathbf{K}} \) are used. \( \Phi \) is operator of the discrete Fourier transform, i.e. \( \Phi(\mathbf{K}) = \hat{\mathbf{K}}. \)
To simplify equation (16), we use the fact that matrix $\Omega$ is Hermitian, i.e. it is equal to its own conjugate transpose\(^\text{12}\). Then it can be represented in the form:

$$\Omega = \mathbf{P}\mathbf{A}\mathbf{P}^\dagger, \quad \Lambda_{ij} = \omega_j^2 \delta_{ij},$$  

(18)

where $\omega_j^2, j = 1, \ldots, N$ are eigenvalues of matrix $\Omega$ and $\omega_j(k)$ are branches of dispersion relation for the lattice; $\dagger$ stands for complex conjugate; matrix\(^\text{13}\) $\mathbf{P}$ is composed of normalized eigenvectors of matrix $\Omega$. Eigenvectors of the dynamical matrix are referred to as polarization vectors \([12]\). Examples of matrix $\mathbf{P}$ are given by formulas (41), (53).

We substitute formula (18) into (16). Then decoupled system of equations with respect to $\mathbf{K}' = \mathbf{P}^\dagger \dot{\mathbf{K}}\mathbf{P}$ is obtained

$$\ddot{\mathbf{K}}'' + 2(\Lambda \ddot{\mathbf{K}}' + \dddot{\mathbf{K}}'\Lambda) + \Lambda^2 \mathbf{K}' - 2\Lambda \mathbf{K}' \Lambda + \mathbf{K}'\Lambda^2 = 0 \iff \dddot{\mathbf{K}}''_{ij} + 2(\omega_i^2 + \omega_j^2) \dddot{\mathbf{K}}'_{ij} + (\omega_i^2 - \omega_j^2)^2 \mathbf{K}'_{ij} = 0.$$  

(19)

Initial conditions for $\mathbf{K}'$ are derived by multiplying formulas (17) by $\mathbf{P}^\dagger$ from the left and by $\mathbf{P}$ from the right. Solving equations (19) with corresponding initial conditions and using the relation (8) for matrices $\mathbf{T}$ and $\mathbf{K}$, yields:

$$\mathbf{T} = \int_k \mathbf{P}\mathbf{T}'\mathbf{P}^\dagger dk, \quad T'_{ij} = \frac{1}{2}\{\mathbf{P}^\dagger \mathbf{T}_0 \mathbf{P}\}_{ij} [\cos((\omega_i - \omega_j)t) + \cos((\omega_i + \omega_j)t)].$$  

(20)

Here $\{\ldots\}_{ij}$ is element $i, j$ of the matrix. Here and below integration is carried out with respect to dimensionless components of the wave vector (see formula (15)). Formula (20) yields an exact expression for the temperature matrix at any moment in time.

If initial kinetic energy is equally distributed between degrees of freedom of the unit cell, then $\mathbf{T}_0 = T_0\mathbf{E}$ and formula (20) reduces to

$$\mathbf{T} = \frac{T_0}{2} \left( \mathbf{E} + \int_k \mathbf{P}\mathbf{B}(t)\mathbf{P}^\dagger dk \right), \quad B_{ij}(t) = \cos(2\omega_j t)\delta_{ij},$$  

(21)

where $\mathbf{E}$ is the identity matrix, i.e. $E_{ij} = \delta_{ij}$. Formula (21) shows that during approach to thermal equilibrium temperature matrix is generally not isotropic\(^\text{14}\), i.e. temperatures, corresponding to degrees of freedom of the unit cell, are generally different even if their initial values are equal.

Thus formula (20) exactly describes time evolution of the temperature matrix. The majority of further results follow from formula (20).

### 3.3 Time evolution of kinetic temperature

In this section, using formula (20) we describe the evolution of the kinetic temperature, $T$, defined by formula (6). According to formulas (5), (6), the kinetic temperature is proportional to the total kinetic energy of the unit cell. Since the total energy per unit cell is conserved, then the evolution of the kinetic temperature is caused by redistribution of energy among kinetic and potential forms.

\(^\text{12}\)Proof of this statement is given in Appendix II.

\(^\text{13}\)Matrix $\mathbf{P}$ is unitary, i.e. $\mathbf{P}\mathbf{P}^\dagger = \mathbf{E}$, where $\mathbf{E}$ is identity matrix, i.e. $E_{ij} = \delta_{ij}$

\(^\text{14}\)Matrix is called isotropic if it is diagonal and all elements on the diagonal are equal.
Kinetic temperature is calculated using formula (20):\(^{15}\)

\[
T = \frac{T_0}{2} \left[ 1 + \frac{1}{N} \sum_{j=1}^{N} \int_{k} \left( 1 + \frac{\{P^*\top \text{dev}T_0P\}^{jj}}{T_0} \right) \cos (2\omega_j(k)t) \, dk \right],
\]

(22)

where \(\text{dev}T_0 = T_0 - T_0E\), \(T_0 = \frac{1}{N} \text{tr}T_0\), \(E\) is identity matrix. Formula (22) shows that evolution of kinetic temperature is influenced by initial distribution of kinetic energy among degrees of freedom of the unit cell. Corresponding example is given in figure 5.

If initial kinetic energy is equally distributed among degrees of freedom of the unit cell then \(\text{dev}T_0 = 0\) and formula (22) reduces to

\[
T = \frac{T_0}{2} \left[ 1 + \frac{1}{N} \sum_{j=1}^{N} \int_{k} \cos (2\omega_j(k)t) \, dk \right].
\]

(23)

This expression can also be derived by calculating trace of both parts in formula (21).

**Remark.** Formula (23) is valid for both monoatomic and polyatomic lattices. It generalizes results obtained in papers [2, 37, 40, 41] for several one-dimensional and two-dimensional monoatomic lattices. For monoatomic scalar lattices \((N = 1)\), formula (23) reduces to the expression obtained in paper [42].

Integrands in formulas (22), (23) are rapidly oscillating functions, frequently changing sign inside the integration domain. Integrals of this type usually tend to zero as time tends to infinity [16]. Therefore kinetic temperature tends to \(\frac{T_0}{2}\). Decrease of temperature is caused by redistribution of energy among kinetic and potential forms. Note that this redistribution is irreversible\(^{16}\).

**Remark.** Investigation of asymptotic behavior of integrals (22), (23) at large times is not a trivial problem. However, using general results obtained using the stationary phase method [16], we can assume that the value \(T - T_0/2\) tends to zero in time as \(1/t^{d/2}\), where \(d\) is space dimensionality. Confirmation of this assumption for some particular lattices is given in papers [2, 37, 58]. Rigorous derivation is beyond the scope of the present paper.

**Remark.** Formulas (20), (22), (23) can be generalized for the case of a finite crystal under periodic boundary conditions. In this case, integrals corresponding to inverse discrete Fourier transform are replaced by sums.

Thus time evolution of the temperature matrix, \(T\), during approach to thermal equilibrium is exactly described by formula (20). The approach is accompanied by two processes: oscillations of kinetic temperature, caused by equilibration of kinetic and potential energies (formulas (22), (23)) and redistribution of kinetic energy between degrees of freedom of the unit cell. From mathematical point of view, the first process is associated with changes of \(\text{tr}T\), while the second process causes evolution of \(\text{dev}T\).

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\(^{15}\)Here the identity \(\text{tr}(PT^*P^{*\top}) = \text{tr}T^\top\) was used.

\(^{16}\)Calculation of change of entropy corresponding to redistribution of energy between kinetic and potential forms would be an interesting extension of the present work.
4 Thermal equilibrium. Non-equipartition theorem

4.1 Random initial velocities and zero displacements

In this section, we show that the temperature matrix tends to some equilibrium value constant in time. Therefore the notion of thermal equilibrium is used. A formula relating equilibrium value of temperature matrix with initial conditions is derived using exact solution (20).

We rewrite formula (20) in the form

\[
T = \frac{1}{2} \int_k P \text{diag} (P^T T_0 P) P^T dk + \int_k P T P^T dk, \\
\hat{T}_{ij} = \frac{1}{2} \{P^T T_0 P\}_{ij} [(1 - \delta_{ij}) \cos((\omega_i - \omega_j) t) + \cos((\omega_i + \omega_j) t)].
\]

Here diag (\ldots) stands for diagonal part of a matrix. The first term in formula (24) is independent of time. In the second term, the integrand is a rapidly oscillating function. Such integrals usually asymptotically tend to zero as \( t \to \infty \) (see e.g. paper [16]). Therefore the second term vanishes at large times. Then the temperature matrix tends to equilibrium value, given by the first term. In order to simplify further analysis, we represent matrix \( T_0 \) as a sum of isotropic part and deviator. Then the first term in formula (24) reads

\[
T_{eq} = \frac{1}{2N} \text{tr} (T_0) E + \frac{1}{2} \int_k P \text{diag} (P^T \text{dev} T_0 P) P^T dk.
\]

Formula (25) relates the equilibrium temperature matrix with initial conditions. It shows that, in general, equilibrium temperatures, corresponding to different degrees of freedom of the unit cell, are not equal. Formula (25) is a particular case of the non-equipartition theorem formulated in the next section.

Formula (25) shows that if initial kinetic temperatures, corresponding to different degrees of freedom of the unit cell, are equal (\( \text{dev} T_0 = 0 \)) then they are also equal at equilibrium (\( \text{dev} T_{eq} = 0 \)). Note that during approach to equilibrium the temperature matrix is generally not isotropic, i.e. \( \text{dev} T \neq 0 \) (see formula (21)).

4.2 Arbitrary initial conditions

In this section, we generalize the results obtained in the previous section for the case of arbitrary initial conditions.

We define generalized potential energy, \( \Pi \), generalized Hamiltonian, \( H \), and generalized Lagrangian, \( L \) as

\[
H = K + \Pi, \quad L = K - \Pi, \quad \Pi = -\frac{1}{4} (\mathcal{L} D + D \mathcal{L}), \\
D(x - y) = M^{\frac{1}{2}} \left( u(x) u(y)^T \right) M^{\frac{1}{2}}.
\]

In papers [37, 41] similar values are introduced for monoatomic lattices.

Consider relation between generalized kinetic and potential energies at thermal equilibrium. In Appendix I, it is shown that covariance of particle displacements, \( D \), and
generalized Lagrangian satisfy the identity

\[ L = \frac{1}{4} \dot{D}. \]  

(27)

We assume that at thermal equilibrium the second time derivative in the right side of formula (27) is equal to zero. Then equilibrium values of generalized kinetic and potential energies are equal

\[ K_{eq} = \Pi_{eq} = \frac{1}{2} H_{eq}. \]  

(28)

Consider a system of equations for equilibrium value of the generalized Hamiltonian, \( H_{eq} \). In Appendix III, it is shown that \( H \) satisfies additional conservation laws. Writing the conservation laws for \( H_{eq} \), yields

\[ \text{tr} H_{eq} = \text{tr} H_0, \quad \text{tr} (L'' \text{dev} H_{eq}) = \text{tr} (L'' \text{dev} H_0), \quad n = 1, 2, \ldots \]  

(29)

Here \( H_0 \) is the initial value of the generalized Hamiltonian. Also \( \text{dev} H \) satisfies equation (10) (see Appendix I). We seek for stationary solution, \( \text{dev} H_{eq} \), of equation (10), formulated for \( \text{dev} H \). Then removing time derivatives in this equation yields:

\[ L^2 \text{dev} H_{eq} - 2L \text{dev} H_{eq} L + \text{dev} H_{eq} L^2 = 0. \]  

(30)

Solution of equations (29), (30) yields equilibrium value of the generalized Hamiltonian, \( H_{eq} \). Given known \( H_{eq} \), other generalized energies and temperature matrix are calculated using formulas (8) and (28).

**Remark.** A system of equations, similar to (29), (30), for crystals with monoatomic lattice and interactions of the nearest neighbors was derived in paper [41]. However solution of this system was obtained only for square and triangular lattices. Here we derive a general solution of system (29), (30), for any polyatomic lattice.

Equations (29), (30) are solved as follows. Applying the discrete Fourier transform (14) to these equations and using formula (18) for matrix \( \Omega \), yields

\[ \Lambda^2 H' - 2\Lambda H' \Lambda + H' \Lambda^2 = 0, \quad \text{tr} (\Lambda^n H') = \text{tr} (\Omega^n \text{dev} \hat{H}_0), \quad H' = P^* \text{dev} \hat{H}_{eq} P. \]  

(31)

Rewriting the first equation from (31) in a component form it can be shown that off-diagonal elements of matrix \( H' \) are equal to zero. The second equation from (31) is represented as

\[ \sum_{j=1}^{N} \omega_j^{2n} H'_{jj} = \sum_{j=1}^{N} \omega_j^{2n} \{P^* \text{dev} \hat{H}_0 P\}_{jj}. \]  

(32)

Then the solution of equations (31) takes form:

\[ H'_{ij} = \{P^* \text{dev} \hat{H}_0 P\}_{ij} \delta_{ij}. \]  

(33)

Substitution of formula (33) into the last formula from (31) allows to calculate matrix \( \hat{H} \). Applying the inverse discrete Fourier transform and using formulas (8), (28), yields:

\[ k_B T_{eq} = \frac{1}{N} \text{tr} (H_0) E + \int_k \text{diag} (P^* \text{dev} \hat{H}_0 P) P^* d\mathbf{k}. \]  

(34)
In the first term, $H_0$ is calculated at $x = y$.

**Remark.** Evolution of the generalized Hamiltonian is governed by the forth order equation (10). Therefore $H_{eq}$ and $T_{eq}$, in principle, can be influenced by $H_0, \dot{H}_0, \ddot{H}_0$. However formula (34) shows that only $H_0$ matters.

Thus formula (34) is a generalization of formula (25) for the case of arbitrary initial conditions. It shows that, in general, equilibrium temperatures, corresponding to different degrees of freedom of the unit cell, are not equal. Formula (34) can be referred to as the *non-equipartition theorem*. The theorem relates equilibrium temperature matrix with initial conditions.

## 5 Example. Diatomic chain

### 5.1 Equations of motion

Presented theory is applicable to crystals with an arbitrary lattice. In this section, the simplest one-dimensional polyatomic lattice is analyzed.

We consider a diatomic chain with alternating masses $m_1, m_2$ and stiffnesses $c_1, c_2$ (see fig. 2). The chain consists of two *sublattices*, one formed by particles with masses $m_1$ and another formed by particles with masses $m_2$. This model is frequently used as an example of a system with two branches of dispersion relation [12, 39, 54, 62].

![Figure 2: Two unit cells of a diatomic chain with alternating masses and stiffnesses. Particles of different size form two sublattices.](image)

We write equations of motion of the chain in matrix form (3). Elementary cells, containing two particles each, are numbered by index $j$. Position vector of the $j$-th unit cell has form:

$$x_j = a je,$$

where $a$ is a distance between unit cells, $e$ is a unit vector directed along the chain. Each particle has one degree of freedom. Displacements of particles, belonging to the unit cell $j$, form a column

$$u_j = u(x_j) = (u_{1j}, u_{2j})^\top,$$

where $u_{1j}, u_{2j}$ are displacements of particles with masses $m_1$ and $m_2$ respectively. Then equations of motion have form

$$M\ddot{u}_j = C_1 u_{j+1} + C_0 u_j + C_{-1} u_{j-1},$$

$$M = \begin{bmatrix} m_1 & 0 \\ 0 & m_2 \end{bmatrix}, \quad C_0 = \begin{bmatrix} -c_1 - c_2 \\ c_1 - c_2 \end{bmatrix}, \quad C_1 = \begin{bmatrix} 0 & 0 \\ c_2 & 0 \end{bmatrix}. \quad (37)$$
Here $C_{-1} = C_1^\top$.

Initially particles have random velocities and zero displacements. Velocities of particles with masses $m_1, m_2$ are chosen such that initial temperatures $T^0_{11}, T^0_{22}$ of the sublattices are different. Velocities of different sublattices are uncorrelated, i.e. $\langle \dot{u}_{1j} \dot{u}_{2j} \rangle = 0$. Then initial temperature matrix has form

$$T_0 = \begin{bmatrix} T^0_{11} & 0 \\ 0 & T^0_{22} \end{bmatrix}, \quad k_B T^0_{11} = m_1 \langle \dot{u}_{1j}^2 \rangle, \quad k_B T^0_{22} = m_2 \langle \dot{u}_{2j}^2 \rangle. \quad (38)$$

Here velocities are calculated at $t = 0$. Initial temperature distribution is spatially uniform, i.e. $T^0_{11}, T^0_{22}$ are independent on $j$. Further we consider time evolution of temperatures of sublattices, equal to diagonal elements, $T_{11}, T_{22}$, of the temperature matrix.

5.2 Dispersion relation

Evolution of temperature matrix is described by formula (20). In this section, we calculate the dispersion relation and matrix $P$ included in this formula.

We calculate dynamical matrix, $\Omega$, by formula (16). Substituting expressions (37) for matrixes $C_\alpha, \alpha = 0; \pm 1$ into formula (16), we obtain:

$$\Omega = \begin{bmatrix} c_1 + c_2 & -c_1 + c_2 e^{-i p} \\ -c_1 + c_2 e^{i p} & c_1 + c_2 \end{bmatrix}, \quad k = \frac{p}{a} e, \quad (39)$$

where $k$ is a wave vector; $p \in [0; 2\pi]$. Calculation of eigenvalues of matrix $\Omega$, yields the dispersion relation:

$$\omega^2_{1,2}(p) = \frac{\omega^2_{\text{max}}}{2} \left( 1 \pm \sqrt{1 - \frac{16 m_1 m_2 c_1 c_2 \sin^2 \frac{p}{2}}{(m_1 + m_2)^2 (c_1 + c_2)^2}} \right), \quad \omega^2_{\text{max}} = \frac{(c_1 + c_2)(m_1 + m_2)}{m_1 m_2}, \quad (40)$$

where index 1 corresponds to plus sign. Functions $\omega_1(p), \omega_2(p)$ are referred to as optical and acoustic branches of the dispersion relation respectively. Note that $\omega_{1,2}/\omega_{\text{max}}$ equally depend on $m_1/m_2$ and $c_1/c_2$. Branches of dispersion relation for different ratios of stiffnesses are shown in fig. 3.

We calculate matrix $P$ in equation (20). By definition, matrix $P$ consists of normalized eigenvectors of dynamical matrix $\Omega$. Eigenvectors $d_{1,2}$, corresponding to eigenvalues $\omega^2_{1,2}$ (formula (40)), have form:

$$d_{1,2} = \begin{bmatrix} 1 - \frac{m_1}{m_2} \pm \sqrt{\left(1 - \frac{m_1}{m_2}\right)^2 + 4|b|^2 \frac{m_1}{m_2}}; -2b \sqrt{\frac{m_1}{m_2}} \end{bmatrix}^\top, \quad b = \frac{c_1 + c_2 e^{i p}}{c_1 + c_2}. \quad (41)$$

Normalization of vectors $d_{1,2}$ yields columns of matrix $P$.

In the following sections, formulas (40), (41) are employed for description of temperature oscillations and calculation of equilibrium temperatures of sublattices.
5.3 Oscillations of kinetic temperature

In this section, we consider oscillations of kinetic temperature of the unit cell $T = \frac{1}{2}(T_{11} + T_{22})$. The oscillations are caused by equilibration of kinetic and potential energies.

Initially particles have random velocities and zero displacements. Initial kinetic energies (temperatures) of sublattices are equal ($T_{01}^{1} = T_{02}^{2}$). The oscillations of kinetic temperature are described by formula (23). In this case, the formula reads

$$T = T_0 + T_{ac} + T_{op}, \quad T_{ac} = \frac{T_0}{8\pi} \int_0^{2\pi} \cos(2\omega_2(p)t)dp, \quad T_{op} = \frac{T_0}{8\pi} \int_0^{2\pi} \cos(2\omega_1(p)t)dp,$$

where $T_0 = \frac{1}{2}(T_{01}^{1} + T_{02}^{2})$ is initial kinetic temperature; dispersion relation $\omega_j(p), j = 1, 2$ is given by formula (40). Contributions of acoustic and optical branches to temperature oscillations are given by integrals $T_{ac}, T_{op}$.

Integrals in formula (42) are calculated numerically using Riemann sum approximation. Interval of integration is divided into $10^3$ equal segments.

To check formula (42), we compare it with results of numerical solution of lattice dynamics equations (37). In simulations, the chain consists of $5 \cdot 10^5$ particles under periodic boundary conditions. Numerical integration is carried out using symplectic leap-frog integrator with time-step $10^{-3}\tau_{\text{min}}$, where $\tau_{\text{min}} = 2\pi/\omega_{\text{max}}$, $\omega_{\text{max}}$ is defined by formula (40). During the simulation the total kinetic energy of the chain, proportional to kinetic temperature, is calculated. In this case, averaging over realizations is not necessary. Time dependence of temperature for $m_1/m_2 = 4$ is shown in fig. 4A. It is seen that analytical solution (42) practically coincides with results of numerical integration of lattice dynamics equations.
Figure 4: A. Oscillations of kinetic temperature \( m_2 = 4m_1, c_1 = c_2 \). Initial temperatures of sublattices are equal. Analytical solution (42) (solid line), and numerical solution (dots). B. Contribution of acoustic \( T_{ac} \), solid line) and optical \( T_{op} \), dots) branches to oscillations of kinetic temperature \( m_2 = 4m_1, c_1 = c_2 \).

Consider contributions \( T_{ac}, T_{op} \) of two branches of dispersion relation to oscillations of kinetic temperature. Time dependencies of \( T_{ac}, T_{op} \) for \( m_2 = 4m_1, c_1 = c_2 \) are shown in fig. 4B. It is seen that contribution of optical branch has a form of beats (two close frequencies), while contribution of acoustic branch has one main frequency. Using the stationary phase method [16] it can be shown that characteristic frequencies of temperature oscillations belong to frequency spectrum of the chain. Group velocities, corresponding to these frequencies, are equal to zero. Figure 3 shows that group velocity of acoustic waves is equal to zero for \( p = \pi \), and group velocity of optical waves vanishes at \( p = 0, p = \pi \). Then main frequencies of temperature oscillations are the following

\[
\begin{align*}
\omega_1|_{p=0} &= \omega_{\text{max}}, \\
\omega_1|_{p=\pi} &= \frac{\omega_{\text{max}}}{2} \left( 1 + \sqrt{1 - \frac{16m_1m_2c_1c_2}{(m_1+m_2)^2(c_1+c_2)^2}} \right), \\
\omega_2|_{p=\pi} &= \frac{\omega_{\text{max}}}{2} \left( 1 - \sqrt{1 - \frac{16m_1m_2c_1c_2}{(m_1+m_2)^2(c_1+c_2)^2}} \right).
\end{align*}
\]

(43)

At large times, oscillations of kinetic temperature is represented as a sum of three harmonics with frequencies (43) and amplitudes, inversely proportional to \( \sqrt{t} \).

Difference between optical frequencies \( \omega_1|_{p=0} \) and \( \omega_1|_{p=\pi} \) decreases with increasing mass ratio, therefore beats of kinetic temperature are observed (see fig. 4). Note that similar beats of temperature are observed in two-dimensional triangular lattice [58].

Consider influence of ratio between initial temperatures of sublattices \( T_{11}^0, T_{22}^0 \) on temperature oscillations. The oscillations for two different cases, \( T_{11}^0 \neq 0, T_{22}^0 = 0 \) and \( T_{11}^0 = 0, T_{22}^0 \neq 0 \), are shown in fig. 5. It is seen that form of oscillations significantly depends on the ratio between \( T_{11}^0 \) and \( T_{22}^0 \). In both cases, analytical results, obtained using formula (23), practically coincide with numerical solution of lattice dynamics equations.
5.4 Redistribution of temperature between sublattices

In this section, we consider the case when initial temperatures of sublattices are not equal \((T_{11}^0 \neq T_{22}^0)\). Then temperature is redistributed between the sublattices.

Numerical solution of equations of motion (37) shows that difference between temperatures of sublattices, \(T_{11} - T_{22}\), tends to some equilibrium value. For example, behavior of \(T_{11} - T_{22}\) for \(m_2 = 4m_1, c_1 = c_2\) is shown in fig. 6. Two cases \(T_{11}^0 \neq 0, T_{22}^0 = 0\) and \(T_{11}^0 = 0, T_{22}^0 \neq 0\) are considered. It is seen that in both cases difference between temperatures tends to the value \(0.3(T_{11}^0 - T_{22}^0)\), predicted by formula (45). Note that shape of curves for two initial conditions is different. Therefore the process of redistribution of temperature between sublattices depends on ratio between \(T_{11}^0\) and \(T_{22}^0\).

We calculate the difference between temperatures of sublattices at thermal equilibrium using formula (25). Deviator of initial temperature matrix has form

\[
\text{dev} T_0 = \frac{T_{11}^0 - T_{22}^0}{2} I, \quad I = \begin{bmatrix} 1 & 0 \\ 0 & -1 \end{bmatrix}.
\]

Substituting (44) into formula (25), yields:

\[
T_{eq} = \frac{1}{4} (T_{11}^0 + T_{22}^0) E + \frac{T_{11}^0 - T_{22}^0}{4\pi} \int_0^{2\pi} \text{P} \text{diag} \left( \text{P}'^T \text{IP} \right) \text{P}'^T dp.
\]

\footnote{This fact follows form the asymptotic analysis based on the stationary phase method \cite{16}.}
Figure 6: Difference between temperatures of sublattices for $T_{11}^0 \neq 0, T_{22}^0 = 0$ (solid line), and $T_{11}^0 = 0, T_{22}^0 \neq 0$ (dots). Here $m_2 = 4m_1, c_1 = c_2, T_{11}^0, T_{22}^0$ are initial temperatures of sublattices.

Matrix $P$ is given by formula (41). Formula (45) yields equilibrium temperatures of sublattices. Integral in formula (45) is calculated numerically using Riemann sum approximation. Interval of integration is divided into $10^3$ equal segments.

Consider the case of equal masses $m_1 = m_2$. Using formula (41) it can be shown that diagonal elements of matrix $P^*TIP$ are equal to zero. Then from formula (45) it follows that for $m_1 = m_2$ and arbitrary $c_1/c_2$ temperatures of sublattices at thermal equilibrium are equal.

To check formula (45), we compare it with results of numerical solution of lattice dynamics equations (37). The chain consists of $10^4$ particles under periodic boundary conditions. We limit ourselves by the following range of parameters: $m_1/m_2 \in [0; 1]$ and $c_1/c_2 \in [0; 1]$. Numerical integration is carried out with time step $10^{-3}\tau_*$, where $\tau_* = \sqrt{\frac{c_1+c_2}{m_1}}$. Initially particles have random velocities such that one of sublattices has zero temperature. During the simulation temperatures of sublattices are calculated. Equilibrium temperatures are computed by averaging corresponding kinetic energies over time interval $[t_{max}/4; t_{max}]$, where $t_{max}$ is the total simulation time. Reasonable accuracy is achieved for $t_{max} = 10^2\tau_*$.

Equilibrium difference between temperatures of sublattices for different mass and stiffness ratios is shown in fig. 7. It is seen that for any given mass ratio, difference between temperatures decreases with decreasing $c_1/c_2$ and tends to a limiting value corresponding to the case $c_1/c_2 \to 0$. In particular, results for $c_2 = 64c_1$ and $c_2 = 32c_1$ are practically indistinguishable.

Thus for the given system, equilibrium temperatures of sublattices are equal if either 1) initial temperatures are equal $T_{11}^0 = T_{22}^0$ or 2) masses are equal $m_1 = m_2$ and stiffness ratio is arbitrary. In general, equilibrium temperatures of sublattices are different. Their
Figure 7: Difference between equilibrium temperatures of sublattices for a diatomic chain. Here $T_{11}^0, T_{22}^0$ are initial temperatures of sublattices. Curves are calculated using formula (45) for $c_1 = c_2 = 1$ (solid line); $\frac{1}{2}$ (doted line); $\frac{1}{4}$ (short dashed line); $\frac{1}{8}$ (dashed line); $\frac{1}{16}$ (dash-dotted line); $\frac{1}{32}$ (dash-double doted line). Circles correspond to results of numerical integration of equations of motion (37).

values are accurately determined by formula (45).

Remark. For equal stiffnesses $c_1 = c_2$, equation (37) is also valid for transverse vibrations of a stretched diatomic chain. In this case, the stiffness is determined by magnitude of stretching force. Therefore all results obtained in this section can be used in the case of transverse vibrations.

Remark. Our results may serve for better understanding of heat transfer in diatomic chains. In papers [32, 61], stationary heat transfer in diatomic chains connecting two thermal reservoirs with different temperatures was investigated. In paper [9] it was shown that for $m_1 \neq m_2$ temperatures of sublattices were different (temperature profile was not smooth), while in paper [61] temperatures of sublattices for $m_1 = m_2$ and $c_1 \neq c_2$ were practically equal. We suppose that these results can be explained using formula (45), which shows that equilibrium temperatures of sublattices are equal only for $m_1 = m_2$.

6 Example. Graphene lattice (out-of-plane motions)

6.1 Equations of motion

In this section, we consider approach to thermal equilibrium in hexagonal lattice (see fig. 8). Only out-of-plane vibrations are considered. The given model describes out-of-plane vibrations of a stretched graphene sheet [3, 6, 21]. In-plane vibrations can be considered separately, since in harmonic approximation in-plane and out-of-plane vibrations are decoupled.

Elementary cells, containing two particles each, are numbered by a pair of indices $j, k$ (see
Basis vectors \( a_1 \) and \( a_2 \) for graphene have form:

\[
a_1 = \frac{\sqrt{3}a}{2} (i + \sqrt{3}j), \quad a_2 = \frac{\sqrt{3}a}{2} (\sqrt{3}j - i),
\]

where \( i, j \) are Cartesian unit vectors; in fig. 8 vector \( i \) is horizontal. Vector \( a_1 \) connects centers of cells \( j,k \) and \( j+1,k \). Vector \( a_2 \) connects centers of cells \( j,k \) and \( j,k+1 \). Position vector of cell \( j,k \) has form:

\[
x_{j,k} = \sqrt{3}a(je_1 + ke_2), \quad e_1 = \frac{a_1}{|a_1|}, \quad e_2 = \frac{a_2}{|a_2|}.
\]

Each particle has one degree of freedom (displacement normal to lattice plane). Displacements of a unit cell \( j,k \) form a column:

\[
u_{j,k} = u(x_{j,k}) = (u^1_{j,k}, u^2_{j,k})^\top,
\]

where \( u^1_{j,k}, u^2_{j,k} \) are displacements of two sublattices.

Consider equations of motion of unit cell \( j,k \). Each particle is connected with three nearest neighbors by linear springs (solid lines in fig. 8). Equilibrium length of the spring is less than initial distance between particles, i.e. the graphene sheet is uniformly stretched\(^{18}\). Stiffness of the spring, determined by stretching force, is denoted by \( c \). Then equations of motion have form

\[
M \ddot{u}_{j,k} = C_{1} u_{j+1,k} + C_{-1} u_{j-1,k} + C_{0} u_{j,k} + C_{2} u_{j,k+1} + C_{-2} u_{j,k-1},
\]

\[
C_0 = \begin{bmatrix}
-3c & c \\
c & -3c
\end{bmatrix}, \quad C_1 = C_2 = \begin{bmatrix}
0 & 0 \\
0 & 0
\end{bmatrix}, \quad M = \begin{bmatrix}
m & 0 \\
0 & m
\end{bmatrix}.
\]

Here \( C_{-1} = C_1^T \), \( C_{-2} = C_2^T \); \( m \) is particle mass.

Initially particles have random velocities and zero displacements. Velocities are chosen such that initial temperatures of sublattices are different \( (T_{11}^0 \neq T_{22}^0) \). Velocities of different

\(^{18}\)In the absence of stretching, out-of-plane vibrations are essentially nonlinear. Various nonlinear effects in unstrained graphene are considered e.g. in papers [4, 34].
sublattices are uncorrelated, i.e. \( \langle \dot{u}_{j,k}^1 \dot{u}_{j,k}^2 \rangle = 0 \). Then initial temperature matrix, \( T_0 \), has form:

\[
T_0 = \begin{bmatrix}
T_{11}^0 & 0 \\
0 & T_{22}^0
\end{bmatrix}, \quad k_B T_{11}^0 = m \langle (\dot{u}_{j,k}^1)^2 \rangle, \quad k_B T_{22}^0 = m \langle (\dot{u}_{j,k}^2)^2 \rangle.
\]

(50)

Here velocities are calculated at \( t = 0 \). Initial temperature distribution is spatially uniform, i.e. \( T_{11}^0, T_{22}^0 \) are independent on \( j, k \). Further we consider time evolution of temperatures of sublattices, equal to diagonal elements of the temperature matrix \( T_{11}, T_{22} \).

### 6.2 Dispersion relation

Evolution of temperature matrix during approach to thermal equilibrium is described by formula (20). In this section, we calculate the dispersion relation and matrix \( P \) included in this formula.

We calculate dynamical matrix \( \Omega \) using formula (16). Substituting expressions (49) for matrixes \( C_\alpha, \alpha = 0; \pm 1; \pm 2 \) into formula (16), we obtain:

\[
\Omega = \omega^2 \left[ \begin{array}{ccc}
3 & -1 - e^{-ip_1} - e^{-ip_2} \\
-1 - e^{ip_1} - e^{ip_2} & 3
\end{array} \right], \quad p_1 = k \cdot a_1, \quad p_2 = k \cdot a_2,
\]

(51)

where \( k \) is wave-vector; \( \omega^2 = \frac{c}{m} \); \( p_1, p_2 \in [0; 2\pi] \) are dimensionless components of the wave vector.

Eigenvalues \( \omega_1^2, \omega_2^2 \) of matrix \( \Omega \) determine dispersion relation for the lattice. Solution of the eigenvalue problem yields:

\[
\omega_{1,2}^2 = \omega^2 \left( 3 \pm \sqrt{3 + 2 \left( \cos p_1 + \cos p_2 + \cos (p_1 - p_2) \right)} \right),
\]

(52)

where index 1 corresponds to plus sign. Functions \( \omega_1(p_1, p_2), \omega_2(p_1, p_2) \) are referred to as optical and acoustic dispersion surfaces respectively (see fig. 9). Eigenvectors of matrix \( \Omega \)

![Figure 9: Acoustic (\( \omega_2(p_1, p_2)/\omega_*, \text{left} \)) and optical (\( \omega_1(p_1, p_2)/\omega_*, \text{right} \)) dispersion surfaces (52) for out-of-plane vibrations of graphene.](image)

are columns of matrix \( P \):

\[
P = \frac{1}{\sqrt{|b|^2 + b^2}} \begin{bmatrix}
|b| & |b| \\
-b & b
\end{bmatrix}, \quad b = 1 + e^{ip_1} + e^{ip_2}.
\]

(53)
In the following sections, formulas (51), (52), (53) are employed for description of temperature oscillations and calculation of equilibrium temperatures of sublattices.

### 6.3 Oscillations of kinetic temperature

In this section, we consider oscillations of kinetic temperature of the unit cell \( T = \frac{1}{2}(T_{11} + T_{22}) \) in graphene.

In general, the oscillations are described by formula (22). Using formulas (50), (53) it can be shown that diagonal elements of matrix \( P^\top \text{dev} T_0 P \) are equal to zero. Then from formula (22) it follows that temperature oscillations are independent of the ratio between temperatures of sublattices \( T_{011} \) and \( T_{022} \). Then formula (23) can be used:

\[
T = \frac{T_0}{2} + T_{ac} + T_{op}, \quad T_{ac} = \frac{T_0}{16\pi^2} \int_0^{2\pi} \int_0^{2\pi} \cos(2\omega_2(p_1, p_2)t)dp_1dp_2, \\
T_{op} = \frac{T_0}{16\pi^2} \int_0^{2\pi} \int_0^{2\pi} \cos(2\omega_1(p_1, p_2)t)dp_1dp_2,
\]

(54)

where \( T_0 = \frac{1}{2}(T_{011} + T_{022}) \) is initial kinetic temperature; functions \( \omega_1, \omega_2(p_1, p_2) \) are given by formula (52). In further calculations, integrals in formula (54) are evaluated using Riemann sum approximation. Integration area is divided into \( 400 \times 400 \) equal square elements.

To check formula (54), we compare it with results of numerical solution of lattice dynamics equations (49). In our simulations, graphene sheet contains \( 10^3 \times 10^3 \) unit cells under periodic boundary conditions. Numerical integration is carried out with time-step \( 5 \times 10^{-3}\tau_* \), where \( \tau_* = \frac{2\pi}{\omega_*} \). During the simulation the total kinetic energy of the lattice, proportional to kinetic temperature, is calculated. In this case, averaging over realizations is not necessary. Time dependence of temperature is presented in fig. 10A.

Figure 10: A. Oscillations of kinetic temperature in graphene sheet with random initial velocities and zero displacements. Numerical solution of equations of motion (49) (dots) and analytical solution (54) (line). B. Contribution of acoustic (\( T_{ac} \), solid line) and optical (\( T_{op} \), dotted line) dispersion surfaces to temperature oscillations in graphene.
figure shows that formula (54) accurately describes temperature oscillations. Calculations with different initial temperatures of sublattices \( T_{11}^0 \neq T_{22}^0 \) confirm our conclusion that the ratio of these temperatures do not influence the behavior of \( T \).

Contributions of acoustic and optical dispersion surfaces to temperature oscillations are shown in fig. 10B. The contributions are given by integrals \( T_{ac} \) and \( T_{op} \) (formula (54)). It is seen that oscillations corresponding to optical dispersion surface has two main frequencies, while oscillations corresponding to acoustic surface has only one main frequency. The frequencies can be calculated using asymptotic analysis of integrals (54) at large \( t \) using the stationary phase method [16]. This investigation is beyond the scope of the present paper. Similar investigation for two-dimensional triangular lattice is carried out in paper [58].

Thus oscillations of kinetic temperature are accurately described by formula (54). Amplitude of these oscillations decays in time as \( 1/t \). Formula (54) is valid for an arbitrary ratio of initial temperatures of sublattices.

6.4 Redistribution of temperature between sublattices

In this section, we consider redistribution of kinetic temperature between sublattices in graphene in the case \( T_{11}^0 \neq T_{22}^0 \).

Equilibrium temperatures of sublattices are calculated using formula (25). Corresponding expression for initial temperature matrix is given by formula (50). In the previous section, it is mentioned that for graphene, matrix \( P^* \mathbf{dev} T_0 P \) in formula (25) has zero diagonal elements. Then from formula (25) it follows that \( \mathbf{dev} T_{eq} = 0 \), i.e. temperatures of sublattices equilibrate.

To check this fact, consider numerical solution of equations of motion (49). Periodic cell containing \( 10^3 \times 10^3 \) unit cells is used. Initially particles of one sublattice have random velocities, while particles of another sublattice are motionless. Initial displacements are equal to zero. Numerical integration is carried with time step \( 5 \cdot 10^{-3} \tau_s \), where \( \tau_s = 2\pi/\omega_s \). Time evolution of temperature difference, \( T_{11} - T_{22} \), is shown in fig. 11. The figure shows
beats of difference between temperatures of sublattices. The amplitude of beats decays in time as $1/t$.

Thus at large times, temperatures of sublattices in graphene become equal.

7 Conclusions

An analytical description of approach to thermal equilibrium in infinite harmonic crystals with complex (polyatomic) lattice was presented.

Initially the crystal is in a nonequilibrium state such that kinetic and potential energies are not equal. The crystal tends to thermal equilibrium, i.e. to a state in which temperatures, corresponding to different degrees of freedom of the unit cell, are constant in time. Approach to thermal equilibrium is accompanied by oscillations of the temperatures exactly described by formula (20). The oscillations are caused by two physical processes: 1) equilibration of kinetic and potential energies, and 2) redistribution of kinetic energy (temperature) among degrees of freedom of the unit cell. In $d$-dimensional crystal, amplitude of the oscillations decays in time as $1/t^d$.

At large times, kinetic and potential energies equilibrate. Kinetic energy is redistributed between degrees of freedom of the unit cell. Equilibrium values of kinetic temperatures, corresponding to different degrees of freedom of the unit cell, are related with initial conditions by the non-equipartition theorem (formulas (25), (34)). The theorem shows that these kinetic temperatures are equal at thermal equilibrium if their initial values are equal. If initial kinetic temperatures are different then they are usually different at equilibrium, except for some lattices. For example, it is shown that in diatomic chain with alternating stiffnesses (equal masses) and graphene lattice performing out-of-plane motions the equilibrium values of the kinetic temperatures are equal.

Our analytical results are exact in the case of spatially uniform distribution of kinetic temperatures. In the case of nonuniform temperature distribution, ballistic heat transfer should be considered along with transient processes described above. However the heat transfer is much slower than the transient processes [19, 38, 42, 56]. Therefore at small times, the crystal locally almost achieve thermal equilibrium. This almost equilibrium state slowly changes due to ballistic heat transfer. Therefore our results can be used for description of fast local transition to thermal equilibrium in nonuniformly heated crystals.

In the present paper, anharmonic effects were neglected. Anharmonicity leads, in particular, to exchange of energy between normal modes. In this case, temperatures, corresponding to different degrees of freedom of the unit cell, tend to equal equilibrium values. However in papers [5, 41, 49], it is shown that, at least in the case of small anharmonicity, the exchange between normal modes is significantly slower than transient thermal processes described above. Therefore at small times, transient thermal processes are well described by harmonic approximation.


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9 Appendix I. Equation for the generalized energies

In this appendix, we show that generalized kinetic energy \( K \), generalized potential energy \( P \), generalized Hamiltonian \( H \), and generalized Lagrangian \( L \) satisfy differential-difference equation (10).

We introduce matrix, \( Z \), consisting of covariances of particle accelerations:

\[
Z = \frac{1}{2} M^\frac{1}{2} \left\langle \ddot{\mathbf{u}}(\mathbf{x}) \ddot{\mathbf{u}}(\mathbf{y})^T \right\rangle M^\frac{1}{2}.
\]  

(55)

Calculation of the second time derivatives of \( K \) and \( Z \) taking into account equations of motion, yields:

\[
\ddot{K} = \mathcal{L}_x K + K \mathcal{L}_y^T + 2Z, \quad \ddot{Z} = \mathcal{L}_x Z + Z \mathcal{L}_y^T + 2 \mathcal{L}_x K \mathcal{L}_y^T,
\]

\[
\mathcal{L}_x K = \sum_\alpha M^{-\frac{1}{2}} C_\alpha M^{-\frac{1}{2}} K(\mathbf{x} + \mathbf{a}_\alpha, \mathbf{y}), \quad K \mathcal{L}_y^T = \sum_\alpha K(\mathbf{x}, \mathbf{y} - \mathbf{a}_\alpha) M^{-\frac{1}{2}} C_\alpha M^{-\frac{1}{2}}.
\]

(56)

Excluding \( Z \) from this system of equations, we obtain

\[
\dddot{K} - 2 \left( \mathcal{L}_x \dddot{K} + \dddot{K} \mathcal{L}_y^T \right) + \mathcal{L}_x^2 K - 2 \mathcal{L}_x K \mathcal{L}_y^T + \mathcal{L}_y^T K(\mathcal{L}_y^T)^2 = 0,
\]

(57)

where \( \mathcal{L}_x^2 = \mathcal{L}_x \mathcal{L}_x \). Formula (57) exactly determines evolution of generalized kinetic energy for any initial conditions.

We consider initial conditions (4), corresponding to spatially uniform temperature distribution. In this case the following identity is satisfied

\[
K(\mathbf{x}, \mathbf{y}) = K(\mathbf{x} - \mathbf{y}).
\]

Using the identity we show that

\[
\mathcal{L}_x K = \mathcal{L} K, \quad K \mathcal{L}_y^T = K \mathcal{L}, \quad \mathcal{L} K \overset{\text{def}}{=} \sum_\alpha M^{-\frac{1}{2}} C_\alpha M^{-\frac{1}{2}} K(\mathbf{x} - \mathbf{y} + \mathbf{a}_\alpha).
\]

(58)

Substitution of expressions (58) for operators into equation (57), yields equation (10):

\[
\dddot{K} - 2 \left( \mathcal{L} \dddot{K} + \dddot{K} \mathcal{L} \right) + \mathcal{L}_x^2 K - 2 \mathcal{L} K \mathcal{L} + \mathcal{L}^2 = 0.
\]

(59)

To derive equations for \( \Pi, L, \) and \( H \), we consider equation for \( D = M^\frac{1}{2} \left\langle \mathbf{u}(\mathbf{x}) \mathbf{u}(\mathbf{y})^T \right\rangle M^\frac{1}{2} \).

Calculation of the second time derivatives of \( D \) and \( K \), yields

\[
\dddot{K} = \mathcal{L}_x K + K \mathcal{L}_y^T + \mathcal{L}_x D \mathcal{L}_y^T, \quad \dddot{D} = \mathcal{L}_x D + D \mathcal{L}_y^T + 4K.
\]

(60)

Excluding \( K \) from this system, we obtain equation for \( D \):

\[
\dddot{D} - 2 \left( \mathcal{L}_x \dddot{D} + \dddot{D} \mathcal{L}_y^T \right) + \mathcal{L}_x^2 D - 2 \mathcal{L}_x D \mathcal{L}_y^T + D(\mathcal{L}_y^T)^2 = 0.
\]

(61)
Additionally, from formula (60) it also follows that
\[ L = \frac{1}{4} \dot{D}. \]  
(62)
Thus \( K \) and \( D \) satisfy the same linear equation (61). Since generalized Hamiltonian, \( H \), and generalized Lagrangian, \( L \), are linear functions of \( K \) and \( D \), then they satisfy equation (61). Trace of the generalized Hamiltonian is constant in time. Therefore \( \text{dev} H \) also satisfies equation (61).

10 Appendix II. Dispersion relation

In this appendix, the dispersion relation for a lattice, described by equations of motion (3), is derived.

We introduce new variable \( U(x) = M^{\frac{1}{2}} u(x) \), then equation of motion (3) takes the form
\[ \ddot{U}(x) = \sum_{\alpha} M^{-\frac{1}{2}} C_{\alpha} M^{-\frac{1}{2}} U(x + a_{\alpha}), \]  
(63)
where \( M^{-\frac{1}{2}} M^{-\frac{1}{2}} = M^{-1} \).

The dispersion relation is derived by making substitution \( U = A e^{i(\omega t + k \cdot x)} \) in formula (63):
\[ (\Omega - \omega^2 E) A = 0, \quad \Omega = -\sum_{\alpha} M^{-\frac{1}{2}} C_{\alpha} M^{-\frac{1}{2}} e^{i k \cdot a_{\alpha}}, \]  
(64)
where \( E \) is identity matrix; \( \Omega \) is referred to as the dynamical matrix [12]. Formula (64) yields a homogeneous system of linear equations with respect to \( A \). The system has nontrivial solution if the following condition is satisfied:
\[ \det (\Omega(k) - \omega^2 E) = 0. \]  
(65)
Here \( \det(\cdot) \) stands for determinant of a matrix. Solutions of equation (65) are branches of dispersion relation \( \omega_j^2(k), j = 1, .., N \). Note that from mathematical point of view, \( \omega_j^2 \), are eigenvalues of \( \Omega \).

Finally, we show that dynamical matrix is Hermitian, i.e. it is equal to its own conjugate transpose:
\[ \Omega^{\ast \top} = -\sum_{\alpha} M^{-\frac{1}{2}} C_{\alpha}^\top M^{-\frac{1}{2}} e^{-i k \cdot a_{\alpha}} = -\sum_{\alpha} M^{-\frac{1}{2}} C_{-\alpha} M^{-\frac{1}{2}} e^{i k \cdot a_{-\alpha}} = \Omega, \]  
(66)
where identities \( a_{\alpha} = -a_{-\alpha}, C_{-\alpha} = C_{\alpha}^\top \) were used.

11 Appendix III. Additional conservation laws for the generalized Hamiltonian

In this appendix, we show that the generalized Hamiltonian satisfies additional conservation laws.
We introduce generalized potential energy, $\Pi(x, y)$, generalized Hamiltonian, $H(x, y)$, and generalized Lagrangian, $L(x, y)$:

$$H = K + \Pi, \quad L = K - \Pi, \quad \Pi = -\frac{1}{4} \left( \mathcal{L}_x D + D \mathcal{L}_y^T \right),$$  \hspace{1cm} (67)

$$D = M^{\frac{1}{2}} \left\langle u(x)u(y)^T \right\rangle M^{\frac{1}{2}}.$$  \hspace{1cm} (68)

Here $K = K(x, y)$; operators $\mathcal{L}_x, \mathcal{L}_y$ are defined by formula (56).

Calculating time derivative of the generalized Hamiltonian taking into account equations of motion, yields:

$$\dot{H} = \frac{1}{4} \left( \mathcal{L}_x W - W \mathcal{L}_y^T \right), \quad W = M^{\frac{1}{2}} \left\langle u(x)v(y)^T - v(x)u(y)^T \right\rangle M^{\frac{1}{2}}.$$  \hspace{1cm} (69)

In the case of spatially uniform initial conditions $W = W(x - y)$ and $\mathcal{L}_x W = \mathcal{L}W$, $W \mathcal{L}_y^T = W \mathcal{L}$. Then

$$\dot{H} = \frac{1}{4} \left( \mathcal{L}W - W \mathcal{L} \right).$$  \hspace{1cm} (70)

Multiplying equation (69) by $\mathcal{L}^n$, calculating trace and using identity $\text{tr} (AB) = \text{tr} (BA)$, yields conservation laws

$$\text{tr} (\mathcal{L}^n H) = \text{tr} (\mathcal{L}^n H_0), \quad n = 0, 1, 2, ..., \hspace{1cm} (71)$$

where $H_0$ is initial value of the generalized Hamiltonian. For $n = 0$ and $x = y$, formula (70) corresponds to conventional law of energy conservation. Similar conservation laws are derived for a one-dimensional chain in paper [7] and for two- and three-dimensional monoatomic crystals in paper [41].

Formula (70) can be written for trace and deviator of the generalized Hamiltonian:

$$\text{tr} H = \text{tr} H_0, \quad \text{tr} (\mathcal{L}^n \text{dev} H) = \text{tr} (\mathcal{L}^n \text{dev} H_0), \quad n = 0, 1, 2... \hspace{1cm} (71)$$

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