Novel dichloromethane-fermenting bacteria in the Peptococcaceae family

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Abstract
Dichloromethane (DCM; CH₂Cl₂) is a toxic groundwater pollutant that also has a detrimental effect on atmospheric ozone levels. As a dense non-aqueous phase liquid, DCM migrates vertically through groundwater to low redox zones, yet information on anaerobic microbial DCM transformation remains scarce due to a lack of cultured organisms. We report here the characterisation of DCMF, the dominant organism in an anaerobic enrichment culture (DFE) capable of fermenting DCM to the environmentally benign product acetate. Stable carbon isotope experiments demonstrated that the organism assimilated carbon from DCM and bicarbonate via the Wood–Ljungdahl pathway. DCMF is the first anaerobic DCM-degrading population also shown to metabolise non-chlorinated substrates. It appears to be a methylotroph utilising the Wood–Ljungdahl pathway for metabolism of methyl groups from methanol, choline, and glycine betaine. The flux of these substrates from subsurface environments may either directly (DCM, methanol) or indirectly (choline, glycine betaine) affect the climate. Community profiling and cultivation of cohabiting taxa in culture DFE without DCMF suggest that DCMF is the sole organism in this culture responsible for substrate metabolism, while the cohabitants persist via necromass recycling. Genomic and physiological evidence support placement of DCMF in a novel genus within the Peptococcaceae family, ‘Candidatus Formimonas warabiya’.

Introduction
Dichloromethane (DCM, CH₂Cl₂) is one of the most commonly encountered subsurface pollutants in industrial areas [1]. Current global production of DCM exceeds 900 Gg y⁻¹, of which 70% is manufactured by humans [2]. The remaining 30% comes from natural sources including biomass burning, oceanic sources, and geothermal activity [2]. Due to widespread production and use of DCM, both surface and tropospheric levels of this toxic chemical continue to rise [3–6]. Atmospherically, DCM has recently been recognised as a potent greenhouse gas with detrimental effects on ozone [7]. The compound also poses a threat to human health [8, 9] and microbial function [10–12].

Nonetheless, microbial transformation of DCM is an option for remediation of oxic and anoxic environments. Aerobic DCM transformation is catalysed by a DCM dehalogenase found in facultative methylotrophs [13, 14], but the enzyme responsible for anaerobic DCM dechlorination has not yet been identified. DCM is a dense non-aqueous phase liquid that descends through groundwater to low redox zones, and so anaerobic degradation plays a vital role in its removal from contaminated sites. Yet study of anaerobic DCM degradation has been hindered by the limited number of organisms capable of this metabolism. Dehalobacterium formicoaceticum strain DMC is the only isolate [15] and from the handful of enrichment cultures [16–18], only ‘Candidatus Dichloromethanomonas...
elyunquensis’ strain RM has been characterised [19, 20]. Both *D. formicoaceticum* and ‘Ca. Dichloromethanomonas elyunquensis’ are obligate anaerobic DCM-degrading bacteria and have genome sequences available [15, 19, 21, 22]. A combination of genomic, physiological, and proteomic work has demonstrated the central role of the Wood–Ljungdahl pathway (WLP) in DCM metabolism in both organisms, however variations on the pathway result in different end products [15, 20, 23]. *D. formicoaceticum* ferments DCM to formate and acetate in a 2:1 molar ratio [15], whilst ‘Ca. Dichloromethanomonas elyunquensis’ completely mineralises DCM to H₂ and CO₂ [23].

The subject of this work is a new anaerobic DCM-degrading lineage, referred to as “DCMF”, that is the dominant organism in a previously reported non-methanogenic bacterial enrichment culture, designated culture DFE [24]. The community was enriched from an organochlorine-contaminated aquifer near Botany Bay, Sydney, Australia and culture DFE has been maintained on DCM as the sole energy source [24]. The culture was recently subjected to PacBio sequencing and the 6.4 Mb, gap-free DCMF genome encoded a complete WLP [24]. However, genome-based metabolic modelling suggested that DCMF may have a wider substrate repertoire due to the presence of 81 full-length MttB superfamily methyltransferases that could confer ability to utilise methylated amines and glycine/betaine/sarcosine reductase gene clusters [24].

Here, we report that DCMF is the first non-obligate anaerobic DCM-degrading bacterium and characterise its metabolism of DCM, quaternary amines, and methanol, whilst also considering the role of the cohabiting bacteria in culture DFE. Stable carbon isotope labelling was used to determine the fate of DCM carbon and function of the WLP. Based on its genomic and physiological novelty, DCMF is proposed to form a novel genus within the *Peptococcaceae* family. Using contemporary molecular and traditional cultivation techniques, this study represents a thorough and robust characterisation of a novel bacterium despite its presence in a multi-lineage enrichment culture.

**Materials and methods**

**Culture medium**

Culture DFE was grown in anaerobic, defined bicarbonate-buffered mineral salts medium as previously described [24]. To investigate the requirement for exogenous bicarbonate during DCM degradation, cultures were instead buffered with 3-morpholinopropane-1-sulfonic acid (MOPS, 4.2 g l⁻¹), either with or without 4 mM NaHCO₃. To study the metabolic fate of DCM, ¹³C-labelled DCM ([¹³C]DCM, 1 mM) was used. To study the assimilation of inorganic carbon, ¹³C-labelled bicarbonate (NaH¹³CO₃, 5 mM) was added to MOPS-buffered culture medium.

To test alternative growth substrates, DCM was replaced with the following (5 mM unless stated otherwise): carbon monoxide (2 mM), choline chloride, dibromomethane, dimethylglycine, formic acid, H₂, glycine betaine, methanol, sarcosine, syringic acid and trimethylamine. Cultures amended with choline chloride, glycine betaine, and trimethylamine were also amended with the following compounds as electron acceptors (15 mM unless otherwise stated): fumarate (80 mM, tested with trimethylamine only), NaNO₂, NaNO₃, Na₂SO₃ and Na₂SO₄. Acetate, H₂, and lactate were tested as electron donors with Na₂SO₃ and Na₂SO₄ as electron acceptors. Glycine betaine and sarcosine (5 mM) were tested as electron acceptors with H₂ (10 mM) as electron donor.

**Analytical methods**

DCM, dibromomethane, acetate, formate, methanol, and trimethylamine were quantified using a Shimadzu Plus GC-2010 gas chromatograph with flame ionisation detector equipped with a headspace autosampler (PAL LHS2-xt-Shim; Shimadzu, Rydalmere, Australia; Table S1). HCO₃⁻ (as gaseous CO₂) and H₂ were quantified using a Shimadzu GC-2010 gas chromatograph with pulsed discharge detector (Table S1). In all analyses, the inlet temperature was 250 °C, split ratio 1:10, FID temperature 250 °C or PDD temperature 150 °C.

Choline and glycine betaine were quantified using liquid chromatography with tandem mass spectrometry. The Agilent 1200 Series LC (Agilent Technologies, Mulgrave, Australia) was fitted with a Luna C18 (2) column (150 × 4.6 mm, 5 μm; Phenomenex, Lane Cove West, Australia). The mobile phases were 0.5 mM ammonium acetate in water (A) and 100% methanol (B). Samples (5 µl) were eluted with a linear gradient from 95:5 (A:B) to 0:100 (A:B) over 10 min, then held at 0:100 (A:B) for 1 min. The LC was coupled to an Applied Biosystems QTRAP 4000 quadrupole mass spectrometer (SCIEX, Mulgrave, Australia) and electrospray ionisation performed in the positive mode. The machine was operated in multiple reaction monitoring (MRM) mode and the following precursor-product ion transitions were used for quantification: *m/z* 104.0 → 59.0 (choline) and *m/z* 118.0 → 57.7 (glycine betaine).

Labelled and unlabelled acetate, CO₂, and HCO₃⁻ were quantified via GC with triple quadrupole mass spectrometer (GC-TQMS) performed with an Agilent 7890 A GC system (Table S1). The TQMS was operated in MRM mode identifying the following precursor-product ion transitions: *m/z* 43 → 15.2 (unlabelled acetate), *m/z* 44 → 15.1 ([¹³C]acetate → 15.2), *m/z* 57.7 (glycine betaine) → 15.1 ([¹³C]glycine betaine), and *m/z* 118.0 → 57.7 (glycine betaine) → 15.1 ([¹³C]glycine betaine) → 15.2 (trimethylamine).
16S rRNA (for total bacterial quantification) samples. The 16S rRNA gene was amplified (the number of 16S rRNA genes in the genome). Numbers were converted to cell numbers by dividing by four (Australia) and the data was analysed with CFX Maestro. Amplicon reads were processed in QIIME2 [27] using the dada2 pipeline [28]: forward and reverse reads were trimmed and joined, chimeras were filtered out. Alpha diversity was assessed with Shannon’s diversity index and pairwise comparisons made with a Kruskal–Wallis test. A two-dimensional PCA plot was created from the weighted Unifrac distance matrix. Samples were compared by the stage of substrate consumption, as well as timepoint, to account for differing metabolic rates between substrates (Table S3).

Bacterial quantification

Genomic DNA was extracted from 2 ml liquid culture as previously described [24]. DCMF and total bacterial 16S rRNA genes were quantified via quantitative real-time PCR (qPCR) with primers Dcm775/Dcm930 and Eub1048/Eub1194 [26], respectively (Table S2). Standard curves were prepared by making serial tenfold dilutions of plasmid DNA carrying cloned DCMF 16S rRNA or Dehalococcoides sp. 16S rRNA (for total bacterial quantification). Reactions were carried out on a CFX96 thermal cycler (Bio-Rad, Gladesville, Australia) and the data was analysed with CFX Maestro v1.0 software (Bio-Rad). DCMF 16S rRNA gene copy numbers were converted to cell numbers by dividing by four (the number of 16S rRNA genes in the genome).

16S rRNA gene amplicon sequencing

Community profiling was carried out on the above DNA samples. The 16S rRNA gene was amplified with the 515F/806R primer pair with adapters (Table S2). Samples were sequenced with Illumina MiSeq technology by The Hawkesbury Institute for the Environment Next Generation Sequencing Facility. Amplicon reads were processed in QIIME2 [27] using the dada2 pipeline [28]: forward and reverse reads were trimmed and joined, chimeras were removed, and samples were rarefied to the lowest sequencing depth. Taxonomy was assigned to genus level using a Naïve Bayes classifier trained on a full-length 16S rRNA gene SILVA database (release 133) and the lowest 1% abundant reads were filtered out. Alpha diversity was assessed with Shannon’s diversity index and pairwise comparisons made with a Kruskal–Wallis test. A two-dimensional PCA plot was created from the weighted Unifrac distance matrix. Samples were compared by the stage of substrate consumption, as well as timepoint, to account for differing metabolic rates between substrates (Table S3).

Cultivation of DFE cohabitant bacteria

To eliminate DCMF and enrich the cohabiting bacteria in culture DFE, two rounds of dilution to extinction cultures (20 ml) were set up in 30 ml glass serum bottles (Fig. S1). These were prepared with the standard medium amended with one of: casamino acids (5 g l\(^{-1}\)), ethanol (10 mM), glucose (10 mM), peptone (5 g l\(^{-1}\)), 1-propanol (10 mM), yeast extract (5 g l\(^{-1}\)). Following qPCR confirmation that the DCMF 16S rRNA gene was below the limit of detection in the lowest active dilution culture, these cultures were subject to Illumina 16 rRNA gene amplicon sequencing and used to inoculate triplicate microcosms amended with one of: 1 mM DCM, 5 mM choline chloride, or 5 mM glycine betaine (Fig. S1), which were monitored for 8 weeks.

Fluorescence in situ hybridisation microscopy

Fluorescence in situ hybridisation (FISH) was carried out with a DCMF-specific oligonucleotide probe (Dcm623, 5′/Cy3/CTCAAGTGCCATCTCCGA-3′), designed using ARB [29], and probe Eub338i (5′/6-FAM/GCTGCCCTCCC GTAGGAGT-3′) [30] to target all bacteria. FISH was carried out as per an established protocol for fixation on a polycarbonate membrane, using minimal volumes of reagents [31]. Cells were fixed with protocols for both Gram-negative [30] and Gram-positive cell walls [32]. Hybridisation was carried out with a formamide-free buffer. Cells were counterstained with VECTASHIELD® AntiFade Mounting Medium containing 1.5 µg ml\(^{-1}\) 4,6-diamidino-2-phenylindole (Vector Laboratories, Burlingame, CA, USA). Images were captured on a BX61 microscope equipped with a DP80 camera (Olympus Australia, Notting Hill, Australia) using Olympus cellSens Dimension software v2.1. DCMF cell length and width was determined from a sample of 20 cells using the linear measurement tool within the programme.

Phylogenetic analysis of DCMF

Whole genome taxonomic analysis of DCMF was carried out with the GTDB-Tk (Genome Taxonomy Database toolkit) [33]. The average nucleotide identity (ANI) tool from the Kostas lab [34] was used to calculate ANI values between DCMF and D. formicoaceticum strain DMC. CompareM (https://github.com/dparks1134/CompareM) was used to calculate the two-way average amino acid identity (AAI) between the anaerobic DCM degraders and other related bacteria in the family Peptococaceae.

Results

Dichloromethane fermentation

After a 21 day lag period, culture DFE consumed DCM at a rate of 133 ± 3 µM day\(^{-1}\), yielding 2.0 ± 1.2 × 10\(^{10}\) DCMF cells per mole of substrate consumed (Fig. 1A). The product...
of DCM fermentation was acetate (0.76 ± 0.03 mol per mole DCM consumed), which was not observed in abiotic controls. DCM attenuation did not proceed in MOPS-buffered cultures free of bicarbonate (Fig. 1B). However, in analogous cultures amended with 4 mM NaHCO₃, DCM attenuation was observed, yet HCO₃⁻ concentrations did not significantly change (p = 0.11, two-tailed t-test between days 0 and 65; Fig. 1B).

**Metabolism of methanol and quaternary amines**

Of the additional substrates tested as sole energy source or with an electron acceptor, DCMF growth was observed when methanol, choline or glycine betaine (5 mM each) were supplied (Fig. 2). Culture DFE consumed methanol at a rate of 309 ± 31 µM day⁻¹ after a 14 day lag period, yielding 0.74 ± 0.04 mol acetate and 5.7 ± 1.4 × 10¹⁴ DCMF cells per mole substrate utilised (Fig. 2A). No methanol depletion was observed in the abiotic (cell-free) control.

Culture DFE consumed choline at a rate of 344 ± 68 µM day⁻¹ after an 11 day lag period, producing 3.1 ± 0.13 mol acetate and 1.3 ± 0.24 mol monomethylamine per mole choline utilised (Fig. 2B). The culture also consumed glycine betaine at a rate of 328 ± 26 µM day⁻¹ after a 7 day lag period, producing 2.3 ± 0.08 mol acetate and 0.95 ± 0.13 mol monomethylamine per mole glycine betaine utilised (Fig. 2C). Trimethylamine, dimethylamine, sarcosine (methylglycine), and glycine were not detectable throughout. Neither acetate nor monomethylamine were detected in abiotic controls, and the latter was also absent from cultures amended with DCM. DCMF cell proliferation aligned with the consumption of these two substrates, yielding an increase 3.0 ± 0.9 × 10¹⁴ cells per mole of choline and 1.1 ± 0.1 × 10¹⁴ cells per mole of glycine betaine utilised (Fig. 2B, C).

DFE cultures amended with quaternary amine metabolic pathway intermediates dimethylglycine and sarcosine (+H₂) also demonstrated production of acetate and monomethylamine, which again aligned with DCMF cell proliferation (Fig. S2). Sarcosine was not degraded in the absence of H₂ (data not shown). Following the observation of DCMF growth and monomethylamine production in cultures amended with sarcosine +H₂, DFE cultures were also set up with glycine betaine +H₂ to determine whether glycine betaine could be reductively cleaved to trimethylamine and acetate. These cultures consumed all glycine betaine (4.4 ± 0.4 mM) and hydrogen (7.9 ± 0.9 mM) within 28 days, producing 15 ± 0.6 mM acetate and 5.5 ± 0.6 mM monomethylamine, but no trimethylamine (Fig. 2D). DCMF cell yields (4.0 ± 2.8 × 10⁸ cells ml⁻¹) were similar to that when glycine betaine was the sole energy source.

Culture DFE was unable to utilise CO, dibromomethane, ethanol, formic acid, syringic acid or trimethylamine as sole energy sources (no growth and/or acetogenesis observed). DCMF was further unable to use any of the tested pairs of electron donors (acetate, choline, glycine betaine, H₂, lactate, trimethylamine) and acceptors (CO₂, fumarate, Na₂SO₄, Na₂SO₃, NaNO₂, and NaNO₃).

**Shifts in DFE community composition in response to substrate consumption**

Community profiling with Illumina 16S rRNA gene amplicon sequencing showed that culture DFE is composed of a limited number of taxa—only 12 amplicon sequencing variants (ASVs) were present at ≥2% relative abundance in >1 sample (Fig. 3). Community composition was similar
in cultures amended with DCM, choline, and glycine betaine, which had a common, DCM-amended inoculum (Fig. 3A–C), but was simplified in cultures that had been maintained on methanol for two sub-cultivations and had a methanol-amended inoculum (Fig. 3D; Fig. S3A). While DCMF was the dominant organism at the time of inoculation and during substrate consumption, its relative abundance decreased markedly in the lag phase prior to substrate consumption, falling to as little as 0.96% in a methanol-amended replicate at day 14 (Fig. 3). Taxa such as Synergistaceae (except in methanol-amended cultures, where this taxon was absent), Desulfovibrio and Veillonellaceae increased in relative abundance during this lag phase, while Spirochaetaceae and Lentimicrobiaceae increased towards the end of and following substrate depletion, particularly in quaternary amine-amended cultures (Fig. 3).

Differences in the DFE community were driven by the degree of substrate consumption (defined in Table S3), more than the various substrates (Fig. S3B). While there was no significant difference in the Shannon diversity index between the samples when grouped by substrate (Kruskal–Wallis p value 0.0976; Fig. S3C), there was a highly significant difference between all groups when clustered by degree of substrate consumption (Kruskal–Wallis p value <0.00001; Fig. S3D).

Exclusion of cohabitants as DCM and quaternary amine consumers

Attempts to isolate DCMF proved unsuccessful [24]. Therefore, to test the hypothesis that DCMF was the sole consumer of DCM and quaternary amines, the cohabiting bacteria in

Fig. 2 Degradation of methanol and quaternary amines by culture DFE. DCMF growth correlated with the depletion of methanol and formation of acetate (A) and the depletion of choline (B) and glycine betaine (C) with formation of acetate and monomethylamine. Cultures amended with glycine betaine and hydrogen (D) did not produce trimethylamine, rather acetate and monomethylamine were once again the products. Substrate and product concentrations are quantified on the left y-axis; DCMF and total bacterial 16S rRNA gene copies are quantified on the right y-axis. Error bars represent standard deviation, n = 3.
culture DFE were enriched to the exclusion of DCMF. This was achieved by dilution to extinction cultures on rich media amended with casamino acids, glucose, peptone, or yeast extract. These growth conditions variously enriched *Bacillus*, *Desulfovibrio*, *Geobacter*, *Petrimonas*, and *Veillonellaceae*, but not DCMF (Fig. S3A–D). *Spirochaetaceae* and *Synergistaceae* phytotypes did not grow on the tested rich media.

The community composition of the four DCMF-free enrichment cultures was considerably altered from the typical, DCM-amended DFE culture, as 16S rRNA gene amplicon sequencing revealed only one to five ASVs in each of the former, compared to >10 in the latter (Fig. S4A–D).

The DCMF-free cohabitant cultures were then tested for their ability to utilise DCM, choline, and glycine betaine.

**Fig. 3** Temporal shifts in the community composition of culture DFE with different substrates. Illumina 16S rRNA amplicon sequencing was used to determine DFE community composition (left y-axis) at timepoints across the growth experiments amended with A DCM, B choline, C glycine betaine, and D methanol reported in Figs. 1 and 2. ASVs are reported down to genus level where possible, otherwise taxonomic level is indicated in the legend ([F] = family, [P] = phylum, [C] = class, [O] = order). Reads with <1% abundance were filtered out in QIIME2. Unassigned reads and ASVs consistently <2% relative abundance were classed together as ‘Other’. Substrate concentration (black circles, right y-axis) and a line connecting the mean substrate concentration at each time point is overlaid on the community composition graphs. These are aligned with the time points written on the x-axis, not drawn to scale.

**Fig. 4** Morphology of DCMF. FISH microscopy images show DCMF cells stained red with the Cy3-labelled Dcm623 probe (A), all bacterial cells stained green with the 6-FAM-labelled Eub338i probe (B), and the overlay of Cy3- and 6-FAM-labelling in these images (C). The scale bars represent 10 µM.
There was no significant substrate depletion in these cultures (Fig. S4E–H), and therefore no evidence of DCM, choline, or glycine betaine degradation by the Bacillus, Desulfovibrio, Geobacter, Petrimonas, or Veillonellaceae phylotypes in culture DFE.

**DCMF morphology**

FISH microscopy enabled selective visualisation of DCMF cells, which appeared rod-shaped and occurred singly or in chains (Fig. 4A). On average, DCMF cells were 1.69 ± 0.27 µm long and 0.64 ± 0.12 µm wide. FISH images confirmed that DCMF numerically dominated culture DFE during DCM dechlorination (Fig. 4C), congruent with community profiling results.

**DCMF is mixotrophic**

To ascertain the fate of DCM carbon, triplicate DFE cultures were amended with [13C]DCM. When 2700 ± 328 µM DCM had been consumed, 666 ± 160 µM of acetate was produced (Fig. 5A), of which 47.1 ± 5.5% was unlabelled, 30.4 ± 2.8% was methyl group labelled ([2-13C]acetate), and 22.5 ± 4.3% was both methyl and carboxyl group labelled ([1,2-13C]acetate; Fig. 5B). A 13C mass balance was achieved by summing the measured concentrations of 13C-labelled carbon in acetate (670 ± 289 µM) and 13C in biomass (994 ± 121 µM; Fig. 5C, Table S4). This amounted to 128 ± 8.2% recovery of the labelled carbon, indicating no unknown fate of DCM in culture DFE.

Analogous work was then carried out with unlabelled DCM in MOPS-buffered medium amended with 13C-labelled bicarbonate, showing that DCMF incorporated carbon from CO2 into the carboxyl group of acetate. The culture consumed 2000 µM DCM and 2150 ± 492 µM 13C from bicarbonate. It produced 973 ± 140 µM acetate (Fig. 5D), of which 45.0 ± 2.3% was unlabelled, 43.5 ± 1.8% was labelled on the carboxyl group ([1-13C]acetate), 2.2 ± 1.3% was labelled on the methyl group, and 9.3 ± 0.1% was labelled on both carbons (Fig. 5E). A mass balance indicated 84.5 ± 7.0% recovery of...
Table 1: Average amino acid (AAI) identity table of DCMF and related bacteria from the Peptococcaceae family.

| Taxon                                      | 1   | 2   | 3   | 4   | 5   | 6   | 7   | 8   | 9   | 10  | 11  | 12  |
|--------------------------------------------|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| DCMF*                                      | 100.0 |     |     |     |     |     |     |     |     |     |     |     |
| Dehalobacterium formicoaceticum DMC*       | 66.5  | 100.0 |     |     |     |     |     |     |     |     |     |     |
| Thermococcales potens DSM 8271             | 55.2  | 54.7 | 100.0 |     |     |     |     |     |     |     |     |     |
| Desulfoarberia orientis DSM 765            | 54.1  | 53.8 | 53.6 | 100.0 |     |     |     |     |     |     |     |     |
| Desulfotomaculum nigrificans CO-1-SRB      | 54.0  | 54.8 | 57.8 | 53.4 | 100.0 |     |     |     |     |     |     |     |
| Pelotomaculum thermopropionicum SI         | 54.0  | 53.3 | 58.3 | 52.0 | 61.1 | 100.0 |     |     |     |     |     |     |
| Desulfococcus acetoxogenes DSM 771         | 54.0  | 53.3 | 56.3 | 53.1 | 59.5 | 60.1 | 100.0 |     |     |     |     |     |
| Desulfobacterium hafniense DCMF            | 53.4  | 54.5 | 53.2 | 63.6 | 53.2 | 52.1 | 52.4 | 100.0 |     |     |     |     |
| Dehalobacter sp. CF                        | 53.3  | 56.1 | 52.5 | 58.2 | 52.0 | 51.5 | 52.6 | 58.5 | 100.0 |     |     |     |
| ‘Candidatus Dichloromethanomonas elyunquensis’ RM* | 53.1  | 53.1 | 53.2 | 59.5 | 52.5 | 52.4 | 52.2 | 59.2 | 70.0 | 100.0 |     |
| Desulfobacterium metallireducens DSM 15288 | 53.1  | 53.4 | 53.4 | 65.1 | 53.2 | 52.9 | 52.6 | 68.1 | 58.5 | 59.5 | 100.0 |
| Syntrophobatina glycolica DSM 8271          | 52.5  | 53.7 | 52.5 | 57.9 | 52.1 | 51.7 | 52.1 | 58.3 | 62.8 | 63.8 | 58.1 | 100.0 |

Bacteria are listed in order of highest to lowest AAI to DCMF and only the species with the highest AAI value was taken from each genus. Dichloromethane-degrading bacteria are marked with an asterisk (*). Numbers in the header correspond to the taxa in the left-hand column.

Phylogenetic analysis of DCMF

Whole genome taxonomic analysis of DCMF with the GTDB-Tk identified its closest relative as Dehalobacterium formicoaceticum strain DMC, placing them together in the novel family Dehalobacteriaceae (order Dehalobacteriales, class Dehalobacteria, phylum Firmicutes). The GTDB re-classified a wide range of bacterial taxa based on its analysis pipeline, including splitting the traditional class of Clostridia (which includes the family Peptococcaceae) into a variety of more specific, monophyletic classes [33], and hence this classification is equivalent to the assignation of family Peptococcaceae previously suggested for DCMF [24]. DCMF had 77.19% ANI to its closest relative, D. formicoaceticum strain DMC. Given that ANI offers robust resolution primarily above 80% values [61], AAI analysis was instead carried out to evaluate genomic distance between DCMF and its closest relatives. D. formicoaceticum was again confirmed as the closest relative to DCMF (AAI value 66.54%), with ‘Ca. Dichloromethanomonas elyunquensis’ and other members of the Peptococcaceae all considerably lower (Table 1).

Discussion

The DFE community

DCMF is a novel candidate species present in enrichment culture DFE, which has been maintained with DCM as sole external energy source for 5 years and at least 20 consecutive transfers [24]. Of the five cohabiting phylotypes previously reported in culture DFE based on 16S rRNA genes identified from PacBio sequencing data [24], four remained amongst the most abundant in the present work (Desulfovibrio, Lentimicrobiaceae, Spirochaetaceae, and Synergistaceae), while one was no longer detected (Ignavibacteria). In combination with the similar community profiles observed across four different substrates, this suggests that culture DFE is a long-term stable-state community.

Illumina amplicon sequencing, FISH microscopy, and qPCR all supported the previous observation [24] of DCMF as the dominant organism in culture DFE during substrate consumption, and linked growth of DCMF to depletion of DCM, methanol, choline, and glycine betaine. Attempts to generate an axenic culture of DCMF have been unsuccessful, similar to the DCM-mineralising bacterium ‘Ca. Dichloromethanomonas elyunquensis’ in culture RM [18, 23]. How the cohabiting organisms in both cultures persist despite numerous transfers and addition of only a simple chlorinated compound (DCM) to minimal, anaerobic medium is a question of interest. While hydrogenotrophic acetogens and methanogens form major sub-populations in culture RM [18, 23], culture DFE is non-methanogenic [24] and was unable to grow on H2 + CO2 alone.

Five of the 12 taxa in culture DFE were categorically excluded from being primary metabolisers of DCM, choline, and glycine betaine when tested in the absence of DCMF (Fig. S4). Spirochaetaceae and Synergistaceae phylotypes, which comprised a significant proportion of the DCM- and methanol-amended communities at various timepoints, respectively, could not be enriched to the exclusion of DCMF. However, their relative abundance

the labelled carbon in acetate (600 ± 84.9 µM), the remaining H13CO3− (2280 ± 170 µM) and 13CO2 (2740 ± 204 µM), and DCMF biomass (710 ± 9.74 µM; Fig. 5F, Table S4).
during growth on all substrates diminished relative to DCMF, suggesting that it is unlikely they are primary consumers of these substrates. Instead, the timing of the changes in relative abundance and known substrate range of major phylotypes in culture DFE suggest that the cohabiting bacteria use cellular detritus resulting from expired DCMF cells as an energy source (i.e. necromass fermentation; Supplementary Discussion).

The role of the WLP in DCM metabolism

Amongst anaerobic DCM-dechlorinating bacteria, DCMF is unique in producing solely acetate as an end product (Fig. 1A). *D. formicoaceticum* produced formate and acetate in a 2:1 molar ratio [15], while ‘Ca. Dichloromethanomonas elyunquensis’ completely mineralised DCM to H₂, CO₂, and Cl⁻ [23]. The latter organism is unique in also encoding and expressing reductive dehalogenases during growth with DCM [19, 20]. Despite these differences, both organisms utilise the WLP for DCM metabolism [15, 20, 23] as is likely the case with DCMF. Removal of bicarbonate from the culture medium precluded DCM dechlorination and ensuing work with ¹³C-labelled DCM and bicarbonate demonstrated that DCMF is mixotrophic, i.e. assimilates carbon from both DCM and CO₂, similar to *D. formicoaceticum* [23].

These experiments also provided compelling evidence for the transformation of DCM to a WLP intermediate, most likely methylene-tetrahydrofolate (CH₂=FH₂; Eq. 1), as has previously been demonstrated in cell-free extracts of *D. formicoaceticum* [35]. In culture DFE, the production of H¹³CO₃⁻ from [¹³C]DCM suggested that CH₂=FH₂ is disproportionated into the WLP where it is oxidised to HCO₃⁻ (Eq. 2, Fig. 6, Table S5). The electrons released could then reduce the remaining CH₂=FH₂ into the methyl group of acetate (Eq. 3). However, the production of unlabelled acetate (47%) indicates that the excess unlabelled HCO₃⁻ (30 mM) in the medium is an alternative electron acceptor to CH₂=FH₂ for acetogenesis (Eq. 4; Fig. 6). The reduction of HCO₃⁻ to acetate requires twice as many electrons for acetate synthesis than CH₂=FH₂ (i.e. eight vs. four). Taking this ratio into account, along with ~1:1 ratio of unlabelled to labelled acetate suggests that ~67% of electrons derived from DCM oxidation were directed toward HCO₃⁻ reduction and 33% to CH₂=FH₂ reduction. Acetogenesis from both HCO₃⁻ and CH₂=FH₂ reduction may allow for greater carbon assimilation and energy production, contributing to DCMF cell yields one
order of magnitude higher than those previously reported for the anaerobic DCM degraders *D. formicoaceticum* and *Ca. Dichloromethanomonas elyunquensis* [23].

\[
4\text{CH}_2\text{Cl}_2 + 4\text{FH}_4 \rightarrow 4\text{CH}_2=\text{FH}_2 + 8\text{H}^+ + 8\text{Cl}^- \quad (1)
\]

\[
3\text{CH}_2=\text{FH}_2 + 9\text{H}_2\text{O} \rightarrow 3\text{HCO}_3^- + 12\text{e}^- + 3\text{FH}_4 + 15\text{H}^+ \quad (2)
\]

\[
\text{CH}_2=\text{FH}_2 + 4\text{e}^- + \text{HCO}_3^- + 4\text{H}^+ \rightarrow \text{CH}_3\text{COO}^- + \text{H}_2\text{O} + \text{FH}_4 \quad (3)
\]

\[
2\text{HCO}_3^- + 9\text{H}^+ + 8\text{e}^- \rightarrow \text{CH}_3\text{COO}^- + 4\text{H}_2\text{O} \quad (4)
\]

The production of [1,2-13C]acetate from [13C]DCM is consistent with the reduction of H13CO3− outlined above. However, the proportion (22.5%) was surprisingly high, given the relatively small contribution that labelled H13CO3− from 2.7 mM [13C]DCM would make to the 30 mM unlabelled HCO3− present in the culture medium. It is possible that co-localisation of WLP proteins in the cytoplasm may cause the reduction of H13CO3− at a higher ratio than expected (i.e. 9%). Studies with [13C]DCM in *D. formicoaceticum* detected the 13C label solely in the methyl group of acetate ([2-13C]acetate), congruent with DCM oxidation stopping at formate [23, 35], while studies with another *Dehalobacterium* species in mixed culture that was capable of formate oxidation similarly detected [1,2-13C]acetate [36].

DFE cultures amended with unlabelled DCM and 13C-labelled HCO3− in MOPS-buffered medium produced an analogous proportion of [1-13C]acetate. A similar proportion of acetate (45.0%) to that observed in the [13C]DCM work was unlabelled, in this case evidently formed using unlabelled HCO3− produced from DCM. Thus, the 13C-labelling experiments support the hypothesis that DCM metabolism involves the WLP and are consistent with the oxidation of formate to HCO3−. As an exogenous supply of formate was unable to stimulate growth of culture DFE, DCMF alone is likely responsible for formate metabolism, which contrasts with the inability of *D. formicoaceticum* to further transform this metabolite [15]. The production of HCO3− from formate balances with its uptake during acetogenesis, congruent with a net flux of approximately zero, leading to the proposal that DCM is transformed as per Eq. 5.

\[
2\text{CH}_2\text{Cl}_2 + 2\text{H}_2\text{O} \rightarrow \text{CH}_3\text{COO}^- + 5\text{H}^+ + 4\text{Cl}^- \quad (5)
\]

**Metabolism of non-chlorinated substrates**

DCMF is the first anaerobic DCM-degrading bacterium that is also capable of metabolising non-halogenated substrates. A genome-based metabolic model previously suggested that the abundance of MttB superfamily methyltransferases (named for their founding member, a trimethylamine:corrinoid methyltransferase) encoded by DCMF may permit growth on methylated amines and/or glycines [24]. While DCMF was unable to metabolise trimethylamine, growth was observed with glycine betaine and the closely related compound choline. Both compounds are quaternary amines with significant environmental roles. Glycine betaine is an osmoprotectant widely used by bacteria [37–39], marine algae [40], marine invertebrates [41], plants [42], and some vertebrates [43]. It is also an important source of nitrogen, comprising up to 20% of the total nitrogen in hypersaline environments [44]. Choline is typically more abundant, albeit as a part of larger molecules including eukaryotic phospholipids, and can be converted to glycine betaine by a near ubiquitous pathway in soil and water environments [45].

Accordingly, DCMF encodes both the choline dehydrogenase (Ga0180325_11215) and betaine aldehyde dehydrogenase (Ga0180325_114191) required for this transformation to glycine betaine. Based on the stoichiometry of observed end products, growth on putative pathway intermediates, and genomic information, we propose that DCMF likely stepwise demethylates glycine betaine to dimethylglycine and then sarcosine (methylglycine), which is then reductively cleaved to form acetate (via acetyl-phosphate) and monomethylamine (Supplementary Discussion). The electron equivalents produced from demethylation can be used for additional reduction of CO2 to acetate via the WLP, as well as the reductive cleavage of sarcosine. This combination of demethylation and reductive cleavage has previously only been observed in *Sporomusa* spp. [46, 47] and is a novel metabolic pathway within the *Peptococcaceae* family. A theoretical energy balance of the product formation and DCMF cell yield suggested that no other organisms in culture DFE were involved in quaternary amine metabolism (Supplementary Discussion).

The DCMF genome also encodes a number of methanol methyltransferases, which are likely utilised for transformation of methanol into CH2=FH2 prior to its entry into the WLP and transformation to acetate (Supplementary Discussion, Fig. 6 and Table S5).

**Environmental significance**

The ability of DCMF to utilise choline, glycine betaine, and methanol suggests that its environmental relevance extends beyond DCM contaminated sites. Coastal salt marshes and intertidal mudflats represent significant sources of methane from the demethylation of trimethylamine, which is in turn derived from quaternary amines [48–50]. Both
trimethylamine and methanol are non-competitive methane precursors, which may allow large methanogen populations to develop in environments where sulphate reduction would typically dominate [51, 52]. Indeed, trimethylamine is responsible for 60–90% of methane production in coastal salt marshes and intertidal sediments [49, 51]. The transformation of quaternary amines to monomethylamine by DCMF provides a pathway of lower methanogenic potential that could operate in coastal subsurface environments. DCMF does create acetate as a major end product, which can be utilised by acetoclastic methanogens. However, unlike methylated amines, methanogens have to compete with more thermodynamically favourable processes such as sulphate reduction for this substrate.

Furthermore, DCM has recently also been recognised as a potent greenhouse gas with ozone-depleting potential [7], and oxygenated hydrocarbons such as methanol can influence atmospheric ozone formation through reactions with nitrous oxides [53]. Therefore, although DCM, methanol, and quaternary amines are seemingly disparate substrates, they are closely linked to the atmospheric flux of climate-active gasses from anoxic, subsurface environments. This is both via the direct influence that DCM and methanol can have on ozone, and the indirect influence of quaternary amines on the flux of methylated amines and methane.

Within an environmental context, 16S rRNA sequences closely related to DCMF have previously been identified at an organochlorine-contaminated site in France [54]. The DCMF ASV had a greater relative abundance in samples taken from the more highly contaminated source (up to 9.78%) and plume (up to 14.7%) sampling wells, compared to the plume boundary (<0.01%) [54]. A BLAST search of the NCBI nucleotide (nr) sequence database also revealed closely related sequences (>90% nucleotide identity) to the DCMF 16S rRNA gene from a range of environmental samples including organochlorine-contaminated aquifers, mud volcanoes, wastewater treatment systems, anaerobic bioreactors, mammalian and insect guts (Fig. S5). This indicates that DCMF and similar phylotypes are widely distributed in the environment, with potential to affect DCM dechlorination and flux of climate-active compounds on a larger scale.

**Provisional classification of DCMF as a novel genus and species**

The family Peptococcaceae (order Clostridiales, class Clostridia) is a physiologically and phylogenetically diverse group of Gram-positive anaerobic bacteria [55–57]. Many of the initial members were isolated from human samples, but other taxa from environmental samples have since been added [57]. The physiological diversity of the family includes chemoorganotrophic, chemolithoautotrophic, chemolithoheterotrophic, and species that are an active gasses from anoxic, subsurface environments. This is both via the direct influence that DCM and methanol can have on ozone, and the indirect influence of quaternary amines on the flux of methylated amines and methane.

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**Description of ‘Candidatus Formimonas warabiya’**

(Form.i.mon`as. L. neut. adj. formicum relating to formic acid or, more generally, one-carbon compounds; Gr. fem. n. monas a monad, unit; N.L. fem. n. Formimonus the one-carbon utilising unit; war.a.bi’ya N.L. neut. n. warabiya the Dharawal name for the area between Botany Bay and Bunnerong, honouring the Traditional Custodians of the land where this bacterium was sampled from. Permission was granted from the Dharawal Language Program research group for use of this word as the species name.)

Strictly anaerobic. Utilises DCM, methanol, choline, glycine betaine, dimethylglycine as sole sources of electrons. Can also utilise the electron donor and acceptor pair H2 and sarcosine. Utilises organic (the aforementioned substrates) and inorganic (CO2) carbon sources. The primary product is acetate; monomethylamine is also produced from choline, glycine betaine, dimethylglycine, and sarcosine +H2. Cells are rod shaped (1.69 × 0.27 μm). The source of inoculum was contaminated sediment from the Botany
Sands aquifer, adjacent to Botany Bay, Sydney, Australia. The type material is the finished genome of ‘Candidatus Formimonas warabiya’ strain DCMF, which is 6.44 Mb and has a G + C content of 46.4% (GenBank accession number CP017634.1; IMG genome ID 2718217647).

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Compliance with ethical standards

Conflict of interest The authors declare that they have no conflict of interest.

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