Synthesis of pyrazolyl thiobarbituric acids and their cytotoxic and antimicrobial evaluation

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ABSTRACT

Synthesis, cytotoxic and anti-microbial screening of novel thiobarbituric acid incorporated pyrazole derivatives were performed. Vilsmeier-Haack reaction of different phenyl hydrazones \(1(\text{a-e})\) afforded pyrazole -4-carbaldehydes \(2(\text{a-e})\) in good yields. Knoevenagel condensation of compounds \(2(\text{a-e})\) with thiobarbituric acid gave a series of 5-ylidene derivatives \(3(\text{a-e})\) in reasonable yields. The synthesized compounds were characterized with the help of IR, \(^1\)HNMR and mass spectral data. The compounds were tested for cytotoxic activity against Vero, MCF-7 and HCT-116 cell lines. Among the tested compounds, 5- \(((3-(4-hydroxyphenyl)-1-phenyl-1H-pyrazol-4-yl)methylene)-2-thioxodihydropyrimidine-4,6-(1H,5H)-dione (3d)\) was found to be most active molecule with the activity against both MCF-7 and HCT-116 cell lines with \(IC_{50}\) values of 14.0 µM and 18.12 µM. In anti-microbial screening, none of the compounds exhibited anti-bacterial activity.

Keywords: Cytotoxic activity, Anti- microbial activity, Thiobarbituric acid, Pyrazoles

INTRODUCTION

Pyrazoles are an interesting class of heterocyclic compounds with synthetic versatility and significant biological activities such as analgesic and anti inflammatory (1-4), anti diabetic (5-7), antimicrobial (8-10), antiproliferative (11-14) etc. In recent years, a number of reports have been documented and few patents have been filed on pyrazole and pyrazoline derivatives as potent anticancer agents with B -Raf kinase inhibitor activity. The high profile NSAID, celecoxib contains a pyrazole nucleus having COX -2 inhibitory activity with few gastro intestinal side effects. However, there is an evidence of an increase in cardiovascular injury with its prolonged use and its anti cancer properties are under clinical trials. Barbituric acid and thiobarbituric acid derivatives have anticonvulsant(15), anti HIV(16), antibacterial(17), cyclin dependent kinase- 2 and tyrosine kinase inhibitor activities(18), 5-benzylidene thiobarbituric acid derivatives and 5-benzylidene-4, 6- pyrimidinediones (18) on the other hand, have been reported as novel tyrosine kinase inhibitors and as well as antimicrobial compounds. It is known that a combination of different bioactive fragments with complimentary pharmacophoric functions and with different mechanism of action usually exhibit synergistic effects. Inspired by this information and also in continuation of our research on pyrazoles, in the present work, we made an attempt to combine these two scaffolds into a single molecule in order to obtain more potent antimicrobial and anti cancer agents. Five novel pyrazolyl thiobarbituric acids \(3(\text{a-e})\) were synthesized by condensing 1,3-diaryl pyrazole carbaldehydes with thiobarbituric acid. Vilsmeier-Haack reaction of phenyl hydrazones \(1(\text{a-e})\) afforded pyrazole carbaldehydes \(2(\text{a-e})\). The structures of the synthesized compounds were confirmed on the basis of FTIR, \(^1\)HNMR and mass spectral data. The synthesized compounds were screened for \textit{in vitro} antibacterial and cytotoxic activities.
MATERIALS AND METHODS

All the solvents and chemicals used were of synthetic grade from SD fine chemicals Ltd., E.Merck, NR chemicals Ltd and Aldrich chemicals. Completion of the reactions was monitored by analytical thin layer chromatography (TLC) using E- Merck 0.25 mm silica gel plates. Visualization was accomplished with UV light (256 nm) and iodine chamber. Purification of synthesized compounds was done by re-crystallization process. The purity of the compounds was checked by a single spot in TLC. Melting points were determined in open capillary tubes using ANALAB melting point apparatus and are uncorrected. All the 1H NMR spectra were recorded on AVANCE 300 MHz spectrometer using DMSO-d6 as solvent and tetra methyl saline (TMS) as an internal standard. Chemical shift values are listed in δ scale. The IR spectra were recorded on Schimadzu FTIR spectrophotometer by using 1% potassium bromide discs. Mass spectra of the compounds were recorded on Agilent 6430 triple quadruple LC-MS system and were given in mass units (m/z).

General procedure for synthesis of 3-aryl-1-phenyl-1H-pyrazole-4-carbaldehydes (2a-e)
To an ice cold solution of DMF (0.1 mol), was added phosphorus oxy chloride (0.013 mol) drop-wise and the temperature was maintained below 10°C, since an exothermic reaction takes place. To the mixture, ice-cold solution of phenyl hydrazone (0.01 mol) was added in lots wise with stirring under ice cold condition. After the completion of the addition, the reaction mixture was stirred and refluxed at 60-70°C for 12 hr. Solution was cooled and poured into crushed ice with stirring and neutralized with NaHCO3 solution. The solid obtained was filtered under suction and recrystallized from methanol.

General procedure for synthesis of 3-aryl-1-phenyl-1H-pyrazol-2-thiobarbituric acids (3a-e)
A mixture of 3-aryl-1-phenyl-1H-pyrazol- 4-carbaldehyde (0.5 g, 2 mmol) and thiobarbituric acid (0.4 g, 2 m mol) in glacial acetic acid (20 mL) and 2-3 drops of piperidine was refluxed for 3-4 hr. A solid was separated from the reaction mixture within 15-20 min and the refluxing was continued for 3-4 hr in order to complete the reaction. The reaction mixture was cooled to room temperature, filtered, and washed with ethanol to give the pure product (0.87 g, 90% yield).

BIOLOGICAL ACTIVITY

Cytotoxic activity
All the synthesized compounds were screened for MTT assay against Vero, (Normal cells) MCF-7 and HCT-116 cell lines and the test was performed at Natco Laboratories, Hyderabad. 1x10^5 cells/well were seeded in 100 µl DMEM supplemented with 10% FBS in each well of 96 well microculture plates and incubated for 24 hr at 37°C in a CO2 incubator. After incubation, cells were treated with test compounds 3(a-e) at 100, 50, 25, 12.5, 6.25 µg/ml concentrations for 48 hr. After 48 hr of incubation, media was removed and to each well 10 µl of MTT (5 mg/ml) was added and the plates were further incubated for 4 hrs. Supernatant liquid from each well was carefully removed and formazan crystals were dissolved in 100 µl of DMSO and absorbance was measured at 540 nm wavelength.

Anti-bacterial activity
All the synthesized compounds were evaluated for anti-bacterial activity as per the standard procedures and the bacterial strains used were procured from National Chemical Laboratory, Pune. The activity was performed against two Gram - positive (S.aureus, B.subtilis) & two Gram - negative organisms (P.aeruginosa, E.coli) at concentrations of 1000, 500, 250 and 125 µg/ml using Cup- plate method.

RESULTS AND DISCUSSION

The method for the synthesis of pyrazole thiobarbituric acids shown under scheme 1. Different substituted phenyl hydrazones (1a-e) were subjected to Vilsmeier-Haack reaction under reflux at 60-70°C for 12 hr to give pyrazole-4 carbaldehydes (2a-e). The purity of the compounds was confirmed by a single spot in TLC. In IR spectra, compounds showed carbonyl absorption around 1670-1680 cm⁻¹ and C=N stretching around 1600 cm⁻¹ which indicated the formation of pyrazole carbaldehydes. Moreover, compounds showed mass ion peaks of 100 % intensity corresponding to their molecular weights in mass spectra further confirmed the structures. The condensation of compounds (2a-e) with thiobarbituric acid in glacial acetic acid and few drops of piperidine under reflux for 3-4 hr afforded pyrazolyl thiobarbituric acids in good yields. The confirmation of the structures of the synthesized compounds was done with the help of FTIR, mass and 1HNMR spectral data. Two carbonyl absorptions around 1710 cm⁻¹ and 1660 cm⁻¹ and N-H streching around 3350 cm⁻¹ in IR spectra indicated the formation of the compounds. Further confirmation was obtained from 1HNMR which showed a singlet integrating for two protons around δ 12.4 due to two NH protons, pyrazole proton appeared as singlet at δ 9.8 and aromatic protons and methine proton together appeared in the range of δ 7.4-8.0. Moreover, mass spectra of the compounds showed molecular ion peaks of 100% intensity corresponding to their molecular weights.
**Physical and spectral data**

**5-((1,3-diphenyl-1H-pyrazol-4-yl)methylene)-2-thioxodihydropyrimidine-4,6-(1H,5H)-dione (3a)**
Cream coloured solid, Yield:75%, m.p:233-236°C; IR(KBr)cm⁻¹:3339 (N-H str),3036(Ar C-H str),1708(C=O str),1658(C=O str),1566(C=N str); ¹HNMR(400MHz,CDCl₃+DMSO-d₆):12.43(s,2H,NH of thiobarbituric acid), 9.84(s,1H, pyrazole), 8.18(s,1H,CH=),7.4-7.9(10H,Ar-H); Mass(m/z): 480.47 (M-H); Anal. Calcd . for C₂₀H₁₄N₄O₂S: Calculated C,64.16; H,3.77; N,14.96%; found C,64.13; H,3.75; N,14.94%.

**5-((3-4-chlorophenyl)-1-phenyl-1H-pyrazol-4yl)methylene)-2-thioxodihydropyrimidine-4,6-(1H,5H)-dione (3b)**
White coloured solid, Yield:78%, m.p:224-228°C; IR (KBr) cm⁻¹: 3340 ( N-H str), 3028 (Ar C-H str),1708(C=O str),1658(C=O str),1566(C=N str); ¹HNMR(400MHz, CDCl₃ +DMSO-d₆) :12.45(s,2H,NH of thiobarbituric acid), 9.84(s,1H, pyrazole), 8.18(s,1H,CH=),7.4-7.9(10H,Ar-H); Mass(m/z): 409.1 (M+H); Anal. Calcd. for C₂₀H₁₃ClN₄O₂S, Calculated C,58.75; H,3.20; N,13.70%; found C,58.72; H,3.18; N,13.68%.

**5-((1-phenyl-3-(p-tolyl)-1H-pyrazol-4-yl)methylene)-2-thioxodihydropyrimidine-4,6-(1H,5H)-dione (3c)**
White coloured solid, Yield:80%, m.p:260-265°C; IR (KBr) cm⁻¹: 3349 (N-H str), 3028 (Ar C-H str),1708(C=O str),1660(C=O str),1558(C=N str),759(C- Cl str); ¹HNMR(400MHz, CDCl₃ +DMSO-d₆) :12.45(s,2H,NH of thiobarbituric acid), 9.84(s,1H, pyrazole), 8.18(s,1H,CH=),7.4-7.9(9H,Ar-H); Mass(m/z): 389.1 (M+H); Anal. Calcd. for C₂₁H₁₆N₄O₂S, Calculated C,64.93; H,4.15; N,14.42% ;found C,64.92; H,4.14; N,14.40%.

**5-((3-(4-hydroxyphenyl)-1-phenyl-1H-pyrazol-4-yl)methylene)-2-thioxodihydropyrimidine-4,6-(1H,5H)-dione (3d)**
White coloured solid, Yield:74%, m.p:280-285°C; IR (KBr) cm⁻¹: 3348(N-H str), 3340 (O-H str), 3028 (Ar C-H str),1729(C=O str),1690(C=O str),1596(C=N str); ¹HNMR(400MHz, CDCl₃ +DMSO-d₆) :12.41(s,2H,NH of thiobarbituric acid), 9.84 (s,1H, pyrazole) , 8.18 (s,1H,CH=),7.4-7.9(9H,Ar-H); Mass(m/z): 432.1 (M+H); Anal. Calcd. for C₁₉H₁₁N₄O₃S, Calculated C,61.43; H,3.78; N,15.54%; found C,61.41; H,3.76; N,15.52%.
D₂O); Mass (m/z): 391.1 (M+H); Anal. Calcd. for C₂₀H₁₄N₄O₃S, Calculated C,61.53; H,3.61; N,14.35% ; found C,61.52; H,3.59; N,14.33%.

5-((3-(4-naphalen-2-yl)-1-phenyl-1H-pyrazol-4-yl)methylene)-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (3e)
White coloured solid, Yield:79%, m.p:298-300°C; IR (KBr) cm⁻¹: 3337 (N-H str), 3047 (Ar C-H str), 1735 (C=O str), 1690 (C=O str), 1596 (C=N str); ¹HNMR (400MHz, CDCl₃+DMSO-d₆), 12.41 (s, 2H, NH of thiobarbituric acid), 9.84 (s, 1H, pyrazole) , 8.18 (s, 1H, CH=), 7.4-8.0 (12H, Ar-H); Mass (m/z): 425 (M+H); Anal. Calcd. for C₂₅H₁₆N₄O₂S, Calculated C,67.91; H,3.80; N,13.20% ; found C,67.86; H,3.68; N,13.16%.

CYTOTOXIC ACTIVITY
In vitro cytotoxic activity in cultured cells by MTT assy
All the synthesized compounds were subjected for MTT assy against 3 cancer cell lines MCF-7, HCT-116 and Vero (Normal cells). The screening was carried out at 100, 50, 25, 12.5, 6.25 µg/ml concentrations. The IC₅₀ values of the compounds were recorded and shown in table-1. Compound 3d exhibited highly potent activity against MCF-7 and HCT-116 cell lines in MTT assay at 14.02 µM and 18.12 µM concentrations. Similarly compound 3b also exhibited slightly less activity against both the cell lines .Other compounds displayed the activity in the range of 33.8 µM-82.0 µM concentrations.

Table 1
| Compound code | Cytotoxicity expressed as IC₅₀(µM) in cell lines |
|---------------|-----------------------------------------------|
|               | Vero | MCF-7 | HCT-116 |
| 3a            | 173.4 | 54.1 | 60.8 |
| 3b            | 189.8 | 22.8 | 24.0 |
| 3c            | >400  | 82.0 | 66.0 |
| 3d            | 145.8 | 14.02| 18.12|
| 3e            | 132.0 | 33.8 | 42.0 |
| Doxorubicin   | 3.1   | 1.544| 1.964|

IC₅₀:50% inhibitory concentration after 48 hr of drug treatment and the values are average of three individual experiments.

ANTIBACTERIAL ACTIVITY
All the synthesized compounds were evaluated for anti-bacterial activity against four organisms at concentrations of 1000, 500, 250 and 125 µg/ml using Cup - plate method against Gram- negative organisms (P.aeurognisa & E.coli) and two Gram – positive organisms(B.subtilis & S.aureus).

None of the compounds have shown significant anti-microbial activity against the tested organisms even at 1000 µg/ml concentration.

CONCLUSION
Five novel pyrazolyl thiobarbituric acid derivatives were synthesized by combining pyrazole moiety with thiobarbituric acid scaffold and the synthesized compounds were characterized on the basis of physical and spectral data. In cytotoxic activity screening, compounds with hydroxyl and chloro substitutions on phenyl ring exhibited highly potent activity against MCF-7 and HCT-116 cell lines while the other three compounds were moderately active. To our surprise, none of the compounds exhibited significant anti-microbial activity. Further investigation of such thiobarbituric acid incorporated pyrazole derivatives could be interesting to get more selective anti-cancer agents.

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REFERENCES
[1] K Mogilaiah ; K Vidya ; S Kavitha, Ind. J. Chem. 2009, 48B, 282-285.
[2] D R Nagargoje; A R Ghawalkar; G R Jadhav, Ind. J. Het. Chem. 2008, 18, 53-56.
[3] K Sriswetha; B Madhava Reddy; V Harinadh babu, Med.Chem.Res. 2013, 22, 4886-4892.
[4] A M Youssef; MS White; EB Villanueva, Bioorg. Med. Chem. 2010, 18, 2019-2028.
[5] Om Prakash; Rashmi Pundeera; Pooja Ranjana, Ind. J. Het. Chem. 2009, 48 B, 563-568.
[6] Prasanna A; Datar Sonali; R Jadhav, Lett.Drug.Design.and Discovery. 2010, 11,686-703
[7] G Rainer; Krutgen; Artaneium Forsch; U Klemm, Chemical Abstracts. 1981, 31, 649.
[8] S N Thore; Ashwini kumar Gupta, Ind. J. Chem. 2010, 19, 329-332.
[9] AA Bekhit; HM Ashour; A D Bekhit, Med. Chem. 2009, 5, 103-117.
[10] EbraheemAbdu Musad; Riyaz Mohamed; Bahjat Ali Saeed, Bioorg. Med. Chem. Lett. 2011, 21, 3536-3540.
[11] LV Peng-Cheng; L Zhu Hai; LI Huan-Qiu, Bioorg. Med. Chem. 2010, 18, 4606-4614.
[12] P Pevarello; M G Brasca; R Amici, J. Med. Chem. 2004, 47, 3367.
[13] Ronghui Lin; George Chiu; Yang Yu, Bioorg. Med. Chem. Lett. 2007, 17, 4557-4561.
[14] MS Christodoulou; S Liekens; KM Kasiotis, Bioorg. Med. Chem. 2010, 18, 4338-4350.
[15] A Agarwal; S Lata; KK Saxena, E. J. Med. Chem. 2006, 41, 1223-1229.
[16] Roberta Costia; Roberto Disanto; Marino Artico, Bioorg. Med. Chem. 2004, 12, 199-215.
[17] T Tomasic; N Zidar; V Rupnik, Bioorg. Med. Chem. Lett. 2009, 19, 153–157.
[18] Z Chen; Q Yan; R Cao, Bioorg. Med. Chem. Lett. 2014, 22, 3279–3284.