The effect of different chelating agents on the push-out bond strength of proroot mta and endosequence root repair material

Purpose
This study aimed to evaluate the effects of 17% ethylenediaminetetraacetic acid (EDTA), 7% maleic acid (MA), and 10% citric acid (CA) on the push-out bond strength of ProRooT MTA and Endosequence Root Repair Material (ERRM) putty.

Materials and Methods
Eighty single-rooted extracted human teeth were instrumented to obtain a standardized immature teeth model. Based on the chelating agents tested, the specimens were randomly divided into three experimental groups: Group 1 (17% EDTA), Group 2 (7% MA), Group 3 (10% CA), and Group 4 (Positive Control) (n=20 for each group). Each group was further classified into two subgroups: Group A (ProRoot MTA) and Group B (Endosequence Root Repair Material (ERRM) putty) (n=10 for each subgroup). After irrigation and placement of cements, teeth were stored at 37°C and in 100% humidity for a week. A total of 240 dentine discs (three discs per teeth) were obtained and subjected to push-out assay. Data was analyzed using two-way analysis of variance and Tukey's post hoc t-test.

Results
Both types of chelating agent and calcium silicate-based cement were significantly associated with the push-out bond strength values. The push-out bond strength was significantly less for CA as compared to EDTA or MA. ERRM had higher bond strength values than ProRoot MTA (p<0.05).

Conclusion
The use of chelating agents increased the push-out bond strength of CSC. Regardless of tested chelating agents, ERRM had higher bond strength values than ProRoot MTA.

Keywords: Calcium silicate-based cement; chelating agent; Endosequence Root Repair Material; push-out bond strength
strength (5). Chelating agents are used to remove the smear layer in root canal therapies. However, these agents cause demineralization and structural changes in root canal dentin, thus affecting the bond strength of endodontic materials to root canal dentin (6).

Ethylendiaminetetraacetic acid (EDTA) is a frequently used chelating agent in removing smear layer and stimulates cellular differentiation and tissue formation, and increases the release of growth factors throughout the root canal (7). However, EDTA can cause weakness of the dentin structure in immature teeth of young patients, as it causes erosion in the dentin tubules (8). Because EDTA interferes with the hydration of MTA, it decreases microhardness, bond strength and biocompatibility of MTA (9). As alternatives to EDTA, citric acid (CA) and maleic acid (MA) can be used in endodontic procedures (10). Both agents cause wider opening in the dentin tubules (11), and increase the bond strength, resulting in an increased contact area between root canal dentin and endodontic cement (12). MA is suggested as an alternative to EDTA. It is highly acidic, less toxic than EDTA and has a greater ability to remove the smear layer than EDTA (10). As MA is a slightly organic acid, it is recommended that MA should be used in root canal irrigation at a concentration range of 5-15% (10). CA, another chelating agent, is used in different concentrations (1-50%) to remove the smear layer (13). CA, when used at a concentration of 1%, also presents a smear layer-removal effect similar to EDTA (14). Silveiro et al. (15) reported that 10% CA was effective in removing the smear layer because its pH was close to the neutral pH and was therefore more biocompatible.

Although mineral trioxide aggregate (MTA) is frequently used with numerous applications in endodontics, it has several disadvantages such as staining of the teeth, difficulty in clinical use, and long setting time (16). Because of these disadvantages, researchers are attempting to overcome the limitations of MTA. Recently, Endosequence Root Repair Material (ERRM), a bioceramic material was produced to overcome the disadvantages of MTA (17). It has similar uses like that of MTA and is available in mix-free, ready-to-use paste or injectable pat forms.

Exposure to irrigation solutions during chemomechanical irrigation changes the chemical and mechanical properties of the root canal dentin surface and so evaluating the effect of chelating agents on the bond strength of CSC should be investigated. Therefore, the purpose of this study was to examine the effects of 17% EDTA, 7% MA, and 10% CA on the push-out bond strength of ProRoot MTA and ERRM. The null hypotheses tested were as follows: (1) the chelating agent has no effect on the push-out bond strength of ProRoot MTA and ERRM; and (2) there is no difference between the push-out bond strength values of ProRoot MTA and ERRM.

**Materials and Methods**

**Teeth Selection**

Ethical approval was obtained from the Health Ethics Committee of the University of Cumhuriyet, Sivas, Turkey (ID: 2016-12/08). Based on the data from a pilot study, the values used in the power analysis were based on the following: $\alpha = 0.05$, $\beta = 0.10$, $\gamma - \beta = 0.90$. It was decided to take a sample of 80 teeth. The present study was conducted on 80 single-rooted human teeth freshly extracted due to periodontal problems. The teeth were immersed in NaOCl (Wizard, Ankara, Turkey) for 3 hours and root surfaces were cleaned using a curette. Teeth were stored in 0.1% thymol solution at 4°C under the laboratory procedures. Multidimensional preoperative radiographs were taken to confirm the root curvature as less than 20° and also to confirm the presence of a single, non-complicated root canal.

**Specimen Preparation**

Each tooth was decoronated below the cementoenamel junction using diamond burs and the root lengths were standardized to $15 \pm 1$ mm. Working length (wl) was determined by inserting a no. 10 K file (Dentsply Maillefer, Ballaigues, Switzerland) into each root canal until apically visible and then subtracting one mm from this point. Each root canal was instrumented with nickel titanium rotary Protaper Next files (Dentsply Maillefer, Ballaigues, Switzerland) up to size F5. During each file change, 1 mL of 2.5% NaOCl was applied with a side vented 27-gauge needle (Monoject, Tyco Healthcare, Mettawa, IL, USA) for 1 min. To provide an immature tooth model with a standard intracanal diameter, Peeso reamers (Mani Inc, Tochigi, Japan) from #1 to #6 were used sequentially. Finally, a #6 Peeso reamer protruded one mm beyond the apical foramen (3). Each root canal was irrigated with 5 ml of 5.25% NaOCl for 5 min. Finally, all roots were irrigated with 15 mL of bidistilled water. The root canals were then dried with paper cones (Dentsply, Maillefer, Switzerland).

The specimens were randomly divided into three experimental groups according to the chelating agents tested: Group 1 (17% EDTA (Wizard, Rehber Chemistry, Istanbul, Turkey)), Group 2 (7% MA (Merck Schuchardt, OHG, Hohenburn, Germany)), Group 3 (10% CA (Cumhuriyet University, Faculty of Medicine,Sivas, Turkey) and Group 4 (Positive Control) (n=20 for each group). Each group was further classified into two subgroups with regard to the type of CSC tested: Group A (ProRoot MTA) and Group B (ERRM) (n=10 for each group).

**Irrigation Procedure and Placement of Cements**

Each tooth was irrigated for 4 min and the total chelating agent volume delivered was 20 mL for each canal (18). Continuous irrigation was applied using a special irrigation device (VATEA, ReDent-Nova, Israel) that pumped the irrigants at the rate of 5 mL/min. Further, the chelating agents were removed by rinsing with 10 mL bidistilled water for 2 min. Approximately 4 mm of each type of cement tested (ProRoot MTA and ERRM) was placed in the coronal third of the root canals by using a MTA gun (MAP System, Dentsply Tulsa Dental, OK, USA) and compressed with hand plugs (Dentsply, Maillefer, Switzerland). ProRoot MTA was manually mixed using a metal spatula with a water to powder ratio of 0.33 following the manufacturer’s recommendations. ERRM is premixed by the manufacturer. Each type of cement was gently applied to the dentinal walls with a moistened cotton pellet. The teeth were wrapped with a wet gauze and stored at 37°C and in 100% humidity for a week (2).
The teeth were embedded in acrylic blocks prepared as apical thirds in acrylic. Parallel transverse sections were obtained with a water-cooled low-speed IsoMet diamond saw (Buehler, Lake Bluff, NY, USA) from the coronal to the apical direction (three slices per tooth) (3). A total of 240 dentin slices (approximately 1 mm-thickness) were obtained. The thickness of each slice was measured using digital calipers (Teknikel, Istanbul, Turkey) with an accuracy of 0.001 mm. The homogeneity of the groups in terms of slice thickness was confirmed through analysis of variance (ANOVA) (p>0.05).

A continuous load was applied to the center of the cements tested using a stainless steel cylindrical plunger of one mm in diameter, mounted onto a Lloyd LRX universal testing machine (Lloyd Instruments, Ltd., Fareham, UK). Loading was applied with a speed of 1 mm min⁻¹ from the apical to coronal direction until dislodgement of the cement occurred. All three slices of each tooth were tested using the push-out test and the mean was taken. The push-out bond strength was calculated in megapascals (MPa) by dividing the maximum load at failure (N) by the area of surface adhesion using the formula (19), area = 2πr × h (where π = 3.14, a constant value, r = radius of intraradicular space, and h = slice thickness in mm) (20).

Evaluation of Failure Patterns

After the push-out test, the fracture surfaces of all specimens were examined with a stereomicroscope (Zeiss) under 25× magnification. Photographs of different fracture types of the specimens were obtained with a stereomicroscope-based photographic machine (Canon EOS 1000D, Canon, Inc., Tokyo, Japan). Each sample was classified into one of the following categories: (i) adhesive failure at cement/dentin interface; (ii) cohesive failure within cement, and (iii) mixed failure in both cement and dentin. One representative specimen for each group was examined for scanning electron microscopy (SEM) analysis. Each specimen was dehydrated in graded ethanol series 25%, 50%, 75%, 90% for 25 min and finally in 100% ethanol for 1 h. The specimens were critically point-dried, mounted on aluminum stubs, sputter-coated with gold/palladium and examined with a scanning electron microscope (SEM) (Leo 440 CCD, Leica, Bensheim, Germany).

Discussion

Previous studies examined the effects of various variables on the bond strength of different CSC (2, 3). These studies reported that the success of endodontic treatments was due to the well-adapted coronal restoration as well as the resistance of the repair agents to displacement forces generated during the condensation of permanent restorative materials. CSC are desired to show dislocation resistance to mechanical forces such as occlusion or condensation of restorative materials (2). It has been reported that the physical properties of endodontic cements change after root canal irrigation (21). Also, the removal of the smear layer causes a closer contact.

### Table 1. Push-out Bond Strength Values (MPa), and Distribution of Failure Modes for Each Group

| Chelating Agents | Calcium Silicate-Based Cement | Mode of Failure | n (A/C/M) |
|------------------|-------------------------------|----------------|-----------|
| 17% EDTA         | ProRoot MTA                   | Endosequence RRM |           |
| 4.71 ± 0.84      | 5.17 ± 0.90                 |                | 11 / 34 / 15 |
| 7% Maleic Acid   | 4.75 ± 1.13                  | 5.23 ± 1.20    | 11 / 13 / 36 |
| 10% Citric Acid  | 3.99 ± 0.89                  | 4.50 ± 0.86    | 36 / 12 / 12 |
| Control (no agent) | 3.97 ± 0.70                  | 4.23 ± 0.81    | 15 / 19 / 26 |

Bond strength values are shown as Mean ± SD. Same lower case letter represents no statistical significant difference within each column, and *represents statistical significant difference within each row, verified by two-way Anova and Tukey’s test (P > 0.05). A: adhesive; C: cohesive; M: mix.
Push-out bond strength of calcium silicate cements

between the cement and root canal dentin which is required for optimal adhesion, as a result this allows chemical bonding or micromechanical interlocking.

There are several studies on the effects of various variables, such as the different types of cement, intracanal medicaments (2, 3), placement techniques of cement, and irrigation regimens (21) on the bond strength of CSC. However, there has been limited research focus on the effect of chelating agents on the bond strength of CSC. Based on this information, the effects of 17% EDTA, 7% MA, and 10% CA on bond strength of ProRooT MTA and ERRM were examined in the present study. Both null hypotheses of the study were rejected because both EDTA and MA increased the bond strength values of endodontic cements as compared to CA. In addition, ERRM was found to have higher bond strength values than ProRooT MTA.

There are several methods to test the bond strength (22). In this study, push-out bond strength test was used. This is a commonly used test to measure the bond strength in the root canal (22). Goracci et al. reported that the push-out test better reflects the clinical conditions of the fracture pattern than microshear or microtensile methods, and is more reliable than other tests (22). Not only were there numerous failures in the preparation of the samples in the microtensile test, the observed data in such tests were distributed over a wide range. On the contrary, the method used in our study allows testing of regional differences and reduces premature failure rates as compared to other tests (19).

Irrigation of root canals with chelating agents such as EDTA, MA or CA is recommended to effectively remove the smear layer, (1, 10, 13, 15). However, MA has been shown to be more biocompatible than EDTA (23), with a better smear layer-removal ability in sclerotic root canals (10). MA at a concentration of 7% was used in this study since higher concentrations may cause damage to intertubular dentin as reported previously (24). CA at a recommended concentration of 10% was used in this study. The decalcifying action of 10% CA was found to be double or more than that of 1% CA (15).

The results of the present study can be attributed to various factors. The first is the region where the discs were obtained. In the present study, dentin discs were obtained from the coronal third of the root canal. This is consistent with the study by Ballal et al. (10), in which the authors reported that one minute application of 7% MA was more effective than 17% EDTA in removing the smear layer in the apical third of the root canal system, but not in the middle and coronal third. In addition, no significant difference was found to exist between MA and EDTA with respect to the degree of microhardness as reported by Ballal et al. (25). In contrast, Ulusoy and Gorgul reported that MA had a higher reduction in dentine microhardness as compared to EDTA (26). In our study, the push-out bond strength was found to be significantly less for CA than for EDTA and MA (p<0.05). This is consistent with a previous finding that CA was less effective than EDTA in removal of the smear layer (27).

Figure 1. Representative stereomicroscope (25×) and scanning electron microscope (SEM) (54×) images of failure modes; (A) adhesive failure at cement/dentin interface; (B) cohesive failure within cement, and (C) mixed failure in both cement and dentin.
Secondly, the results of the present study can also be attributed to the irrigation procedure employed in our study. There is no definite protocol of the type or concentration of chelating agents. However, different irrigation solutions have been shown to affect the adhesion of materials to dentine surfaces as a result of the effect on dentinal walls which includes alteration of surface energy or wetting ability of dentinal walls. Consistent with this study, Ballal et al. (10) have reported the decreased surface tension of 17% EDTA compared to 7% MA, which may be a possible explanation for the higher bond strength of tested CSC in our study. In addition, while EDTA has been shown to cause complete demineralization of the root canal dentine, MA and CA generate mineral gradients (10). One reason for the lower bond strength values of CA groups may be because the decalcifying capacity of CA is time-dependent. Lopez et al. (28) reported that the amount of Ca2 extracted in the CA and EDTA solutions increased with longer immersion time. Consistent with the findings of the present study, Ballal et al. reported that MA is highly acidic and has a better demineralizing effect (25).

In this study, irrespective of the chelating agents tested, ERM was found to have higher bond strength values than ProRoot MTA (p <0.05). One of the reasons may be due to the physical and chemical properties of cements that were tested. The presence of zirconium oxide improved certain physical properties of bioceramics. The composition and particle size of existing cements affect the interaction between cement and root canal dentin (21), ERM has a smaller particle size than MTA. In addition, ERM can form chemical bonds with root canal dentin walls, thus creating a robust connection. It was argued that the bioceramic cements when reacted with moisture, form hydroxyapatite that may chemically bond to the tooth structure (29). This may result in a 2% expansion because of the setting reaction, thus adapting better to the root canals. Because of crystal growth in dentin tubules, the effect of dentinal bridge formation can be strengthened, thereby increasing micromechanical involvement. Furthermore, the bond strength of ERM may be higher than that of ProRoot MTA owing to the particle structure and hydrophilic properties of ERM (30). Inconsistent with the present study, Shokohijenad et al. (31) reported that bond strength of MTA and ERM paste was significantly lower in samples stored in gradients (10). One reason for the lower bond strength values of MA groups may be because the decalcifying capacity of MA is time-dependent. Lopez et al. (28) reported that the amount of Ca2 extracted in the CA and EDTA solutions increased with longer immersion time. Consistent with the findings of the present study, Ballal et al. reported that MA is highly acidic and has a better demineralizing effect (25).

Conclusion

Within the limitations, it may be concluded that the use of chelating agents increased the push-out bond strength of CSC. Both of EDTA and MA increased the bond strength of CSC when compared to CA. ERM had higher bond strength values than ProRoot MTA.
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