Observation of tunable electrical bandgap in large-area twisted bilayer graphene synthesized by chemical vapor deposition

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Although there are already many efforts to investigate the electronic structures of twisted bilayer graphene, a definitive conclusion has not yet been reached. In particular, it is still a controversial issue whether a tunable electrical (or transport) bandgap exists in twisted bilayer graphene film until now. Herein, for the first time, it has been demonstrated that a tunable electrical bandgap can be opened in the twisted bilayer graphene by the combination effect of twist and vertical electrical fields. In addition, we have also developed a facile chemical vapor deposition method to synthesize large-area twisted bilayer graphene by introducing decaborane as the cocatalyst for decomposing methane molecules. The growth mechanism is demonstrated to be a defined-seeding and self-limiting process. This work is expected to be beneficial to the fundamental understanding of both the growth mechanism for bilayer graphene on Cu foil and more significantly, the electronic structures of twisted bilayer graphene.

Bilayer graphene (BG) has attracted much attention due to its unique electronic structure, which can be modified by the stacking orders1–3. AA-stacked BG (AA-BG) has a massless Dirac spectrum, which results in the gapless characteristics under external electric fields similar to that of monolayer graphene (MG)4. In comparison, Bernal-stacked BG (AB-BG) has a massive Dirac spectrum, and a tunable bandgap can be opened by applying vertical electric fields which break the layer symmetry5–7. Thus, AB-BG with tunable bandgap is expected to be used for potential applications in digital electronics and photonics. Besides AA- and AB-BG, BG can also be stacked in the twisted sequence, where the twist angel varies from 0° to 60° (AA stacking: 0°; AB stacking: 60°)8–10. For the AB-BG growth reported previously11,12, the twisted BG regions have always observed which can apparently influence the electrical properties of AB-BG. Consequently, the investigation of electrical properties of twisted BG is not only significant for the fundamental research but also the potential applications of BG film.

Although there are several reports about the electronic structures of twisted BG, the undisputed conclusion is still lacked. Early studies suggested that twisted BG has a massless Dirac spectrum similar to that of AA-BG, and a gap cannot be opened by applying external electric fields13–15. However, in the most recent report16, the studies by using angle-resolved photoemission spectroscopy (ARPES) have demonstrated that the slightly twisted AA-BG (0.1°–0.4°) has the coexisting massive and massless Dirac spectra; even in the small twist angle (0.1°), a gap can be opened by the breakage of interlayer-coupling and potential symmetry, which is caused by the combination effect of twist and applied electrical fields.
Therefore, it is significant to investigate whether twisted BG has a tunable electrical (or transport) bandgap under external electric fields.

In the research area of BG, another hot spot is the chemical vapor deposition (CVD) synthesis of BG film on Cu foil, which has advantages in large-area growth and low cost \(^{17}\). Until now, the CVD growths of BG film on Cu foil have been reported by many groups \(^{18-21}\); however, in theses previous reports, the growth conditions were more strict and complicated than those of monolayer graphene, such as long growth time (~3 h)\(^{18}\), finely tuned growth pressure \(^{19}\), and two-step process \(^{20,21}\). Thus, it is desirable to develop a more simple CVD method for BG growth.

Herein, a facile CVD method is proposed to synthesize the twisted BG film on Cu foil, and the growth parameters are same with those of MG film except spin-coating decaborane onto the Cu foil before growth. More significantly, the electrical studies reveal that as-synthesized twisted BG has a tunable electrical bandgap under vertical electrical fields, which has not been observed before. We expect that this work may be beneficial to the fundamental understanding of both the growth mechanism for BG film on Cu foil and more importantly, the electronic structures of twisted BG.

Results

Synthesis of twisted BG film. Twisted BG film was synthesized by a facile CVD method. Firstly, 25-μm-thick Cu foil (99.8%, Alfa Aesar) was cleaned in the HCl/H\(_2\)O (1:10) solution, and then washed by DI water several times. Secondly, decaborane (B\(_{10}\)H\(_{14}\)) employed as the cocatalyst was spin-coated onto the Cu foil, which is critical for the formation of twisted BG film. Thirdly, the Cu foil was loaded into the silica tube of the CVD system with a vacuum background of \(7 \times 10^{-4}\) Pa, and then the growth chamber was heated to 1000°C and held for 20 min with 30 sccm Ar, and then the CH\(_4\)/H\(_2\) (15/30 sccm) replacing Ar were introduced into the tube for graphene growth at 1000°C for 20 min. Finally, cooled the system to room temperature with a cooling rate of 50°C/min in CH\(_4\)/H\(_2\) ambience. For comparison, MG film was synthesized by the same procedure of twisted BG except without spin-coating decaborane onto the Cu foil.

Transmission electron microscopy characterization of twisted BG film. Transmission electron microscopy (TEM) was carried out to characterize the structure of the as-synthesized twisted BG film. Figure 1a shows the high-resolution TEM image of random edge of graphene film, indicating bilayer structure. 8 random selected area electron diffraction (SAED) patterns have been obtained, and all reveal two typical hexagonal crystalline structure of graphene with rotation angles. As shown in Fig. 1b–f and Supplementary Fig. S1, the rotation angle varies from 0 to 30° in different regions (3.8°, 8.2°, 9.0°, 9.2°, 15.6°, 16.6°, 18.9° and 22.7°), which indicates that the twisted BG film has polycrystalline structure with...
varied twisted stacking orders\textsuperscript{11}. It is noted that neither AA- nor AB- stacking orders are observed. It means even if there are some AA-BG or AB-BG regions in the twisted BG, the proportion would be very small.

**Optical characterizations of twisted BG films.** Figure 2a shows the photograph of large-area twisted BG film transferred onto a 285-nm SiO\textsubscript{2}/Si substrate with a size of 1 inch × 1 inch. (b) Transmittances of MG and twisted BG films transferred onto quartz substrates. The inset shows the photograph of MG and twisted BG films transferred onto quartz substrates. (c) Typical Raman spectrum of twisted BG film transferred onto a 285-nm SiO\textsubscript{2}/Si substrate; the inset shows that the 2D peak can be deconvoluted into four peaks. (d) $I_d/I_{2D}$ (e) FWHM of 2D peak and (f) $I_d/I_G$ Raman mappings of twisted BG film at the 100 × 100 μm\textsuperscript{2} scale.

In order to further investigate the distribution of twist angles in the twisted BG film, 50 random regions have been studied by Raman spectra in the area within 1 × 1 cm\textsuperscript{2}. 26 out of 50 regions have the R' Raman bands (~1622 cm\textsuperscript{-1}; see Fig. 2c or the green spectrum in Supplementary Fig. S2a), indicating that the range of twist angle is 5° ~ 10°\textsuperscript{25–28}. 6 out of 50 regions have the R Raman bands (~1489 cm\textsuperscript{-1}; see the blue spectrum in Supplementary Fig. S2a), indicating that the range of twist angle is 10° ~ 14°\textsuperscript{25–28}. Furthermore, according to the FWHM and intensity of 2D band shown in Supplementary Fig. S2b and Fig. S2c, the histogram of twist angles of 50 regions can be finally determined\textsuperscript{25–28}. As shown in Supplementary Fig. S2d, the proportion of twisted angles smaller than 10° is 74%, indicating that grains with small twist angle (<10°) are predominant in the polycrystalline twisted BG film.
Discussion

In the growth process of twisted BG film, decaborane is believed to be as the cocatalyst for promoting the decomposition of methane molecules on the Cu foil, which can be demonstrated as follows. Firstly, for comparative studying, the graphene film was synthesized by using the same procedure of twisted BG except without spin-coating decaborane onto the Cu foil before growth. The high-resolution TEM and Raman spectrum shown in Supplementary Fig. S3a and Fig. S3b reveal that the as-synthesized graphene film is monolayered. It suggests that decaborane is a most critical factor for growing twisted BG film. Furthermore, the effect of decaborane was studied by performing X-ray photoelectron spectroscopy (XPS) measurements. As shown in the Fig. 3a, there is a weak B peak in the high-resolution B1s XPS spectrum of twisted BG film grown on Cu foil; however, the position of B peak is the same as that of decaborane, suggesting that this weak B peak is originated from the residual decaborane on the twisted BG film. Moreover, after transferring the twisted BG film onto a 285-nm SiO₂/Si substrate, any B-related XPS signal is below the detection limit, which implies that residual decaborane can be effectively removed during transferring process. Therefore, the above studies have confirmed that decaborane do not react with graphene film or Cu foil, and only simply plays the cocatalytic role of decomposing methane molecules on the Cu foil during the growth process of twisted BG.

For further understanding the growth mechanism of twisted BG, the effect of growth time on the graphene thickness was investigated. When the growth time is 20 min, the twisted BG film is continuous and fully covers the Cu foil. When the growth time is increased to 120 min, the thickness of graphene film remains unchanged which can be confirmed by the Raman spectra shown in Fig. 3b. Furthermore, the initial stage of growth was studied by setting the growth time at 5 min. As shown in the scanning electron microscopy (SEM) image of Fig. 3c, graphene islands are formed on Cu foil for 5 min growth. From the corresponding Raman spectrum of graphene islands shown in Fig. 3d, the I_G/I_2D intensity ratio and 2D band FWHM value are ~1.0 and ~50 cm⁻¹, respectively, indicating bilayer graphene. It is noted that 10 random islands have been studied by Raman spectra, and all regions show bilayer structures. Thus, we can conclude that the growth of twisted BG is not a layer-by-layer mechanism, and its thickness is defined at the initial stage of growth.

Figure 4 shows the schematic illustration of proposed growth mechanism of twisted BG film synthesized on the Cu foil. At the initial growth stage, methane molecules are catalytically decomposed on the surface of Cu foil with the assistance of decaborane, and subsequently BG islands (seeds) are formed. When the growth time is further increased, BG islands laterally grow with the maintenance of their thickness. Finally, these islands merge into a continuous BG film. This mechanism is most likely a defined-seeding and self-limiting process, which is similar with that of BG film synthesized under a
finely tuned growth pressure\textsuperscript{19}. It is noted that our twisted BG film is formed under the similar growth conditions of monolayer graphene, which is due to the cocatalytic effect of decaborane. It implies that the multilayer graphene may be formed by adopting the more effective cocatalyzer for prompting methane decomposition. In addition, AB stacking region has not been observed in this work. Previous report indicated that high surface diffusion coefficient was beneficial for the formation of AB stacking order with lowest energy configuration\textsuperscript{20}, thus we think that besides being as the cocatalyzer for methane decomposition, decaborane also hinders the diffusion of active carbon species, resulting in the formation of both varied twisted stacking orders and disorder for the twisted BG film\textsuperscript{20,31}.

Electrical transport measurements have been performed to investigate the electronic structures of twisted BG film, in particular to verify the existence of tunable electrical bandgap. Figure 5a shows the schematic structure of dual–gated BG field effect transistor (FET). The BG channel is sandwiched completely between top and bottom gates; 70 nm Al\textsubscript{2}O\textsubscript{3} and 285 nm SiO\textsubscript{2} act as top-gate and bottom-gate dielectric films, respectively. The detailed fabrication procedures are given in the Methods Section. For this dual-gated structure, the electrical bandgap and carrier doping concentration of twisted BG can be independently controlled by the effects of top and bottom gate voltages (\(V_{tg}\) and \(V_{bg}\)), which can be described by the top and bottom electrical displacement fields (\(D_t\) and \(D_b\)). \(D_t\) and \(D_b\) can be calculated using the following equations

\[
D_t = +\frac{\varepsilon_t}{d_t} (V_{tg} - V_{tg}^0) \tag{1}
\]

and

\[
D_b = -\frac{\varepsilon_b}{d_b} (V_{bg} - V_{bg}^0) \tag{2}
\]

respectively. \(\varepsilon_t\) and \(d_t\) are dielectric constant and thickness of top-gate dielectric layer, respectively \(\varepsilon_{Al\textsubscript{2}O\textsubscript{3}}: 7.5\), \(d_{Al\textsubscript{2}O\textsubscript{3}}: 50\) nm); \(\varepsilon_b\) and \(d_b\) are dielectric constant and thickness of bottom-gate dielectric layer, respectively \(\varepsilon_{SiO\textsubscript{2}}: 3.9\), \(d_{SiO\textsubscript{2}}: 285\) nm); \(V_{tg}^0\) and \(V_{bg}^0\) are the Dirac offset top and bottom voltages due to the initial environment-induced carrier doping, respectively\textsuperscript{5}.

Figure 5b shows a two-dimensional contour plot of the device resistance \(R\) versus \(V_{tg}\) and \(V_{bg}\). The values of \(R\) reach maximum at the upper-left and lower-right corners, where average displacement fields (\(D_{ave}\)) are highest. \(D_{ave}\) is defined as...
The variation of $R$ is similar with that of AB-BG, which implies a tunable bandgap opening in twisted BG under vertical electric fields\textsuperscript{5,20,32}. This can be more evidently illustrated in the curves of $R$ as a function of $V_{tg}$ at fixed $V_{bg}$ ranging from $-80$ to $30$ V, with $10$ V increment. For each curve, the Dirac resistance, $R_{\text{Dirac}}$, has the highest value, corresponding to the charge neutrality ($D_t = D_b$). As shown in Fig. 5c, $R_{\text{Dirac}}$ increases with increasing $V_{bg}$ in both positive and negative directions, which implies that the bandgap increases with increasing the electrical displacement fields\textsuperscript{5,20}. Furthermore, because the lower bandgap causes the lower $R_{\text{Dirac}}$, the Dirac point with the lowest $R_{\text{Dirac}}$ can be identified as the zero-gap Dirac point ($D_t = D_b = 0$) and used to estimate the Dirac offset voltage induced by the environmental doping ($V_{tg}^0 = -3.7$ V, $V_{bg}^0 = -10$ V). Consequently, the relationship of $R_{\text{Dirac}}$ versus $D_{\text{ave}}$ can be obtained as shown in Fig. 5d. It shows that the $R_{\text{Dirac}}$ reaches the maximum value at the highest $D_{\text{ave}}$, which further confirms the tunability of bandgap in twisted BG film\textsuperscript{17,20,32}.

$$D_{\text{ave}} = \frac{(D_t + D_b)}{2} \quad (3)$$

30 dual-gated BG FET devices have been measured, and moreover, for comparison, the dual-gated MG FETs have also been fabricated and measured. As shown in Fig. 6a, the electrical studies reveal that 27 out of 30 dual-gated BG FET devices show the tunable electrical bandgap characteristics; the other 3 devices show the typical zero-gap characteristic, which is similar with that for the dual-gated MG FET shown in Supplementary Fig. S4\textsuperscript{32}. Thus, based on the TEM and electrical studies, we can conclude that a tunable electrical bandgap is indeed opened in twisted BG under the combination effect of twist and vertical electrical fields. To our best knowledge, it is the first time to observe the electrical bandgap with tunable characteristics for twisted BG film.

Furthermore, it is noted that previous report indicated that for CVD polycrystalline graphene, the mean grain size was smaller than 2 $\mu$m\textsuperscript{33}; for our polycrystalline twisted BG, the grain size may be further decreased because besides being as the cocatalyst for methane decomposition, decaborane also hinders the diffusion of active carbon species. Thus, we can conclude that the device channel for dual-gated BG FET (4 $\mu$m $\times$ 10 $\mu$m) contains several grains of different twisted stacking orders. It is known that the interlayer coupling increases with decreasing rotation angle for twisted BG; especially, as shown in Fig. 6b, when the rotation angle is small ($<10^\circ$), the AA and AB regions (similar with AA and AB stacking BG) emerge, corresponding to strong interlayer coupling\textsuperscript{14}. Consequently, we think that the tunable electrical bandgap characteristics are originated from grains with small rotation angles ($<10^\circ$), which are predominant in our polycrystalline twisted BG film shown in Supplementary Fig. S2d. Furthermore, our electrical result and previous APRES study\textsuperscript{16} both imply that there is a transition of the tunable bandgap...
with twist angle. Although the exact effect mechanism of twist angle on the electronic structures of twist BG film need to be further investigated in future studies, we think the first observation of tunable electrical bandgap for twisted BG is beneficial for the fundamental understanding of its electronic structures. In addition, the carrier mobilities of twisted BG film extracted from 30 fabricated devices can be estimated to be 672–1695 cm²/V·s by using the well-developed procedures (see the Methods Section)²⁰,³⁴, which can be compared to those of CVD BG films reported previously²¹.

In summary, the twisted BG film has been synthesized on the Cu foil by a facile CVD method by introducing decaborane as the cocatalyst for decomposing methane molecules. Furthermore, the electrical studies reveal that the combination effect of twist and vertical electrical fields can break the interlayer-coupling and potential symmetry, and open a tunable electrical bandgap in twisted BG film. This work is beneficial to the fundamental understanding of not only the growth mechanism for BG film, but also the electronic structures of twisted BG.

**Methods**

**CVD growth of twisted BG film.** Initially, 25-μm-thick Cu foil (99.8%, Alfa Aesar) was cleaned in the dilute HCl/H₂O (1:10) solution for 3 min, and then washed by deionization (DI) water several times to remove the residual acid solution, and then was dried by nitrogen gas. Secondly, decaborane was dissolved into anisole solvent (0.03 g/ml), and then spin-coated onto the surface of Cu foil at 3000 rpm for 30 s. Thirdly, the Cu foil was loaded into the silica tube of the CVD system with a vacuum background of 7×10⁻⁴ Pa, and then the growth chamber was heated to 1000 °C and held for 20 min with 30 sccm Ar, and then the CH₄/H₂ (15/30 sccm) replacing Ar was introduced into the tube for graphene growth at 1000 °C for 20 min. Finally, cooled the system to room temperature with a cooling rate of 50 °C/min in CH₄/H₂ ambience.

**CVD growth of MG film.** Monolayer graphene film was synthesized under the same growth parameters of twisted BG except without spin-coating decaborane onto the Cu foil before growth.

**Transfer process of graphene films.** Firstly, poly(methyl methacrylate) solution (PMMA, A4) was spin-coated on the surface of graphene grown on the Cu foil at 3000 rpm for 30 s, and then the Cu foil was etched away in the FeCl₃ solution for 12 h. Secondly, PMMA/graphene film was rinsed repeatedly using the DI water (10 times), and then transferred to the H₂O/HCl/H₂O₃ (20:1:1) solution for 15 min for removing the residual Cu particles, and then transferred to the H₂O/NH₄OH/H₂O₂ (20:1:1) solution for 15 min for removing the insoluble organic contaminants. Note that the PMMA/graphene film was rinsed using the DI water to remove the residual solution after each cleaning step. Thirdly, the PMMA/graphene was transferred onto the target substrate (SiO₂/Si or quartz substrates), and then was cured at 150 °C for 10 min after natural drying, and then the PMMA was removed using acetone³⁵.

**Characterization.** The high-resolution TEM images and SAED pattern were taken with an FEI Tecnai G2 microscope. The optical transmittances were measured by using a Perkin-Elmer model Lambda 750 UV–vis–NIR spectrophotometer. The Raman spectra were collected with a Renishaw InVia Raman microscope using a 514-nm laser beam (20 mW; 1 cm⁻²). XPS was performed on a Kratos XSAM800 using Al Ka radiation (144 W, 12 mA, 12 kV).

**Fabrication of graphene devices and Electrical measurement.** Dual-gate graphene FET devices were fabricated as follows. Firstly, graphene film (twisted BG or MG) was transferred onto a 285 nm SiO₂/Si substrates. Secondly, source and drain electrodes (Ni/Au: 50/50 nm) were defined and deposited by
using photolithography and e-beam evaporation (285 nm SiO2 and p+ -Si acted as the bottom-gate dielectric and electrode, respectively). Thirdly, the graphene was patterned into strips (4 μm width and 10 μm length) as device channels by using photolithography and Oxygen plasma etching. Fourthly, 50-nm-thick Al2O3 top-gate dielectric film was deposited as the top gate dielectric by atomic layer deposition (ALD). Fifthly, top-gate electrodes were patterned and then metals were evaporated (Ni/Au: 50/50 nm). Finally, the devices were annealed at 250°C for 2h with H2/Ar (100/100 sccm).

Electrical measurements were carried out with an Agilent 4155B semiconductor parameter analyzer in air at room temperature. To extract the field-effect mobility of graphene FET devices, the total resistance of the device, \( R_{\text{total}} \), can be expressed as below:\(^{34}\):

\[
R_{\text{total}} = R_{\text{contact}} + R_{\text{channel}} = R_{\text{contact}} + \frac{L}{W n \mu}
\]

where \( R_{\text{contact}} \) is the contact resistance of the metal/graphene contact; \( R_{\text{channel}} \) is the resistance of graphene channel; \( L \) and \( W \) are the channel and width length, respectively; \( n \) is the carrier concentrations in the graphene channel region; \( e \) is the electron charge; \( \mu \) is carrier mobility. \( n \) can be approximated by

\[
n = \sqrt{n_0^2 + n_{tg}^2} = \sqrt{n_0^2 + (C_{tg} (V_{tg} - V_{\text{Dirac}})/e)^2}
\]

where \( n_0 \) represents the density of carriers at Dirac point; \( n_{tg} \) is top-gate-modulated carrier density; \( C_{tg} \) is the top-gate capacitance \((\varepsilon_0 \varepsilon_{Al2O3}/d_{Al2O3})^{36}\). In order to extract the carrier mobility, we define

\[
y = a + b \times (c + x)^{-1/2}
\]

where \( y = R_{\text{total}} \); \( a = R_{\text{contact}} \); \( b = \frac{L}{W C_{tg} \mu} \); \( c = \frac{n_0^2 \times e^2}{C_{tg}} \); \( x = (V_{tg} - V_{\text{Dirac}})^2 \) to fit the measured data. So the carrier mobility can be obtained by

\[
\mu = \frac{L}{W C_{tg} b}
\]

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Author Contributions
Y.F.C. and P.J.L. proposed and supervised the project. J.B.L., P.J.L. and Y.F.C. discussed and designed the experiments. J.B.L., P.J.L., Z.G.W., F.Q., J.R.H., B.J.Z. and J.H.Z. carried out the experiments. L.G. carried out TEM experiments. P.J.L. and J.B.L. wrote the manuscript. Y.F.C., W.L.Z. and Y.R.L. revised the manuscript. All the authors participated in discussions of the research.

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