Chemotaxonomical and Palynological Studies on Nine Acacia Species Occurring in Sudan

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Abstract

The present study was carried out on the chemotaxonomy and palynology of nine Acacia species occurring in Sudan. It included a preliminary phytochemical screening of the flowers extracts using thin layer chromatography (TLC) followed by isolation of the major compounds from A. nilotica flowers. Palynological results showed that the studied species, except for A. laeta, can be divided into two types; Type 1 pollen characterized by spinescent stipules and Type 2 pollen typified by non-spinescent stipules. A. laeta, in addition to all characters of Type 2 pollen, has polyads with 18-20 monads, a fact supporting its hybrid origin. TLC of extracts revealed the presence of phenolic substances with selective reactivity towards phenolic detecting-reagents. TLC comparison of the isolated compounds with extracts from studied species revealed that gallic acid and its methyl ester were only persistent in Acacia species exhibiting spinescent stipules. Thus, suggesting their use as good chemotaxonomic markers.

Keywords: Acacia, flower compounds, taxonomic markers, gallic acid and its methyl ester, Acacia laeta

1. Introduction

The genus Acacia Miller currently includes about 1300 species and form the second-most species-rich genus in the family Leguminosae (Mabberley, 1997; Maslin, 2001). Acacia is widespread with a number of species in Africa, the Americas, Asia and Australia. Two third of its species are indigenous to Australia which makes Acacia the largest genus of vascular plants in the region (Maslin, 2001). The contribution of morphological, anatomical, ontogenetic and some phytochemical characters has been useful for the classification of the genus Acacia (Bentham, 1842, 1862, 1875; Vassal, 1972, 1981; Evans et al., 1977, 1993; Guinet & Vassal, 1978; Pedley, 1986; Guinet, 1990; Dube et al., 2001; Bouchenak-Khelladi et al., 2010). In spite of this, there is still some controversy regarding the infrageneric taxonomy.

Acacia species (31 species) predominate in Sudan and are of high importance not only because of the gum produced by some of them but also for their medicinal and economical importance. They represent about one third of the African species (Elamin, 1972). The taxonomy of the different species is still not quite clear and several classifications have been proposed. Many botanists had published work related to Sudan Acacia (Crowfoot, 1929; Brown & Massey, 1929; Andrews, 1952; Harison & Jackson, 1958; Elamin, 1972; El Tinay et al., 1979). However, most of the published taxonomical work focused mainly and/or solely on the inflorescence character and classified them into spicate and capitate groups. Elamin (1972) provided the latest and more comprehensive taxonomical account on Sudan Acacja. He rejected the traditional classification which was based on inflorescence type and instead, he used palynology and anatomy in relation to vegetative characters, especially those derived from stipules. He proposed the division of Sudan Acacia into four subgeneric groups:

Group I: Stipules spinescent. Spines at nodes usually white straight and having a vascular system. Inflorescence globose heads, flowers white-cream, yellow, orange. Pollen grains in polyads (16-celled). 2n= 26, 52, 100. Trees and shrubs.

Group II: Stipules non-spinescent. Prickles at nodes in pairs usually dark brown and having no vascular system. Inflorescence cylindrical spikes, pink or white-cream. Pollen grains in polyads (16-celled). 2n= 26. Trees and
shrubs.

Group III: Stipules non-spinenescent. Prickles scattered on internodes and rachis usually dark brown and having no vascular system. Inflorescence spikes and globose heads, flowers white-cream. Pollen grains in polyads (16-celled) smaller in size than in Group II. 2n= 26. All members of this group are climbers.

Group IV: Stipules spinescent, spines at nodes usually white straight and having a vascular system. Inflorescence cylindrical spikes, flowers white-cream. Pollen grains in polyads (28-32) celled). 2n= 26. This Group is typified by only one species; *A. albida* Del. (Now is transferred to the genus *Faidherbia*; *F. albida* (Del.) A. Chev)

The classification of plants by lumping taxa according to morphological similarities is probably the oldest and most widely-used of all the approaches (Quinlan, 1993). However many approaches evolved over time towards the taxonomy of plants. As a result of the limited adequacy of traditional morphological taxonomy, where morphological and anatomical classifications prove complicated in certain instances, other sources of information are sought after. Chemosystematic data are usually valuable in the precise classification within the genera, complementing the other methods for reliable taxonomic classification (Stuessy, 2009). The present study is part of an attempt to subject Sudan *Acacia* to palynological and chemotaxonomical investigation to have a better understanding about the taxonomy of this genus in the light of the recent evidence.

2. Materials and Methods

2.1 Plant Materials

Authenticated flower specimens were collected from nine *Acacia* species (*A. ehrenbergiana* Hayne; *A. laeta* R. Br. ex Benth.; *A. nilotica* (L.) Willd. ex Del. *ssp nilotica* Brenan; *Acacia nilotica* (L.) Delile *subsp. tomentosa* (Benth.) Brenan; *A. mellifera* (Vahl) Benth.; *A. senegal* (L.) Willd.; *A. seyal* var. *seyal* Brenana; *A. sieberana* DC and *A. tortilis* (Forsk.) subsp. *raddiana* (Savi) Brennan growing wild in Khartoum. The botanical vouchers collected were taxonomically identified by late Prof. H.M. Elamin and deposited at Soba Forestry Research Centre Herbarium.

2.2 Phytochemistry

2.2.1 General Experimental Procedures

Thin layer chromatography (TLC) was performed on pre-coated silica gel plates (DC-Alufolien 60 F254, Merck) and preparative TLC (PTLC) plates were home prepared using silica gel H (Merck) powder. Column chromatography (CC) was carried out on polyamide (MN-polyamide SC6 polycaprolactam) and Sephadex LH-20 (Merck) columns. Spots were located by using 2% FeCl3/MeOH and 5% AlCl3/EtOH spraying reagents which are indicative for the presence of phenolics. NMR spectra were performed with a Bruker AM-400 instrument operating at a proton frequency of 400.13 MHz. The electron-impact (70eV) mass spectra were recorded on Varian CH-4 and CH-7 instruments.

2.2.2 Extraction

Flowers (20 g) were air dried at shade and coarsely powdered. The flowers of each species were extracted with aqueous methanol (70%) in a Soxhlet apparatus for 24 hrs. The extracts were filtered and the methanol (MeOH) was evaporated under reduced pressure. The aqueous extracts were fractionated by partition with petroleum ether (PE) and ethyl acetate (EtOAc) successively. The EtOAc extracts were then dried using anhydrous sodium sulphate and evaporated to dryness under reduced pressure. For large scale extraction, the air-dried powdered flowers of *A. nilotica* *ssp* tomentosa (1 kg) were extracted with same above mentioned procedure.

2.2.3 Preliminary Screening

The EtOAc extracts of different flowers were subjected to preliminary screening on TLC using the ascending method. Chromatoplates were developed on solvent system: Toluene: EtOAc: Formic acid (5:4:1) (1), visualized under UV and sprayed with FeCl3/MeOH and AlCl3/EtOH spraying reagents.

2.2.4 Isolation of Pure Compounds

The EtOAc (95 g) extract was chromatographed on a polyamide column eluted with water/MeOH mixtures. A total number of 60 fractions (75 mL each) were collected and finally 8 fractions were obtained on combining the eluates according to their chromatographic similarity behaviour on TLC. Fraction 2 (4 g) was applied to Sephadex LH-20 column using MeOH as eluent to obtain compounds 1 (250 mg) and 2 (300 mg). Fraction 5 (840 mg) was applied to PTLC using PE: EtOAc (8:2) as solvent system to obtain compounds 3 (22 mg) and 4 (97 mg).
2.2 Palynology

Pollen samples were prepared by using the acetolysed technique as described by Erdman (1969). Anthers of flowers were placed in centrifuge tubes, moistured with 5 mL of glacial acetic acid and crushed with a glass rod. The tubes were immediately inverted and left to drain for about 2 hr at room temperature. The pollens were then mounted with glycerine jelly on slides and sealed with colourless nail polish. The prepared slides were examined using a light microscope with an oil immersion objectives (X 100 magnification), dimensions measured and photographs were taken.

3. Results

3.1 Screening of the EtOAc Extracts of Acacia Species

Phytochemical screening of the EtOAc extracts of all studied Acacia species on TLC showed that A. ehrenbergiana, A. nilotica, A. seyal, A. sieberana and A. tortilis reacted positively to FeCl₃ reagent. All species gave violet-coloured dominant spots. In addition, A. nilotica revealed a grey-coloured spot. Moreover, A. tortilis, A. sieberana, A. mellifera, A. laeta and A. senegal responded positively to the treatment with AlCl₃ reagent when detected under UV light (254 nm). All of them exhibited yellow-coloured spots on TLC plates (Figure 1).

![Figure 1. Thin layer chromatography of the ethyl acetate extracts from the flowers of Acacia species. Adsorbent: silica gel; Solvent system: toluene: ethyl acetate: formic acid (5:4:1); Spray reagents: FeCl₃/MeOH and AlCl₃/EtOH](image)

3.2 Characterization of Isolated Compounds from A. nilotica ssp. Tomentosa

3.2.1 Compound 1 (gallic acid methyl ester)

Compound 1 (250 mg) was obtained as brownish powder. Rₐ value 0.68 in solvent system (1), gave violet spot with FeCl₃ reagent. The ¹HNMR spectrum (400 MHz, CD₃OD) was simple, showing only two peaks at δ 3.78 (3H, OCH₃) and 7.04 (2H, H-2 and H-6). The high resolution mass spectrum of compound 1 showed an E⁺ peak at 184 (M⁺) suggesting the molecular formula C₈H₈O₅. On the basis of this spectroscopic evidences and data reported in the literature (Khalid et al., 1988) compound 1 was identified as gallic acid methyl ester (Figure 2).
3.2.2 Compound 2 (gallic acid)

Compound 2 (300 mg) was obtained as greyish powder. Rf value 0.59 in solvent system (1), gave violet spot with FeCl3 reagent. The 1HNMR spectrum (400 MHz, CD3OD) was very simple, showing only a single aromatic singlet at δ 7.04 (2H, H-2 and H-6). The high resolution mass spectrum of compound 2 showed an E+ peak at 170 (M+) suggesting the molecular formula C7H6O5. On the basis of this spectroscopic evidences and comparison with authentic sample compound 2 was identified as gallic acid (Figure 2).

3.2.3 Compound 3 (naringenin)

Compound 3 (22 mg) was obtained as light yellow-coloured powder. Rf value of 0.40 on solvent system (1) and gave dark grey colour after spraying with FeCl3 reagent. The 1HNMR spectrum (400 MHz, C2D6O) exhibited AAXX system centered at δ 7.15 and 7.64 (H-2', 3', 5' and 6'). An ABX system was observed at δ 2.97 (dd, J= 3 and 17 Hz, H-3trans) and 3.42 (dd, J= 13 and 17 Hz, H-3cis). The remaining signals centered at δ 6.20 ((d, J= 2Hz, H-6), 6.21 (d, J= 2 Hz, H-8) and 12.43 (s, 5-OH). The high resolution mass spectrum of compound 3 showed an E+ peak at 273 (M++H) suggesting the molecular formula C15H12O5. On the basis of this spectroscopic evidences and data reported in the literature (Khalid et al., 1988) compound 3 was identified as naringenin (Figure 2).

3.2.4 Compound 4 ((+) catechin)

Compound 4 (97 mg) was obtained as light yellow- coloured powder. Rf value of 0.23 on solvent system (1) and gave violet colour after spraying with FeCl3 reagent. The 1HNMR spectrum (400 MHz, C2D6O) displayed aromatic resonances of three protons in an AMX system centered at δ 6.83 (d, J=2 Hz, H-2'), 6.76 (d, J= 8 Hz, H-5') and 6.71 (dd, J= 2 and 8 Hz, H- 6') and in association with other meta-coupled pair of doublets at δ 5.85 (d, J= 2 Hz, H-6) and 5.39 (d, J= 2Hz, H-8). Four protons signals in the aliphatic region in the form of an ABXY system at δ 4.59 (d, J= 8 Hz, H-2), 3.97 (m, H-3), 2.50 (dd, J= 8 and 16 Hz, H-4') and 2.84 (dd, J= 5 and 16 Hz, H-4'). An ion peak at m/z 289.0718 [M-H] - indicated that the molecular mass was 290 corresponding to the formula C15H14O6. On the basis of this spectroscopic evidences and data reported in the literature (Khalid et al., 1988) compound 4 identified (+) catechin (Figure 2).

Comparison of the EtOAc extracts of the studied Acacia flowers using the isolated compounds as markers in TLC suggests the presence of gallic acid and its methyl ester in A. ehrenbergiana, A. nilotica, A. seyal, A. sieberana, and A. tortilis. The two later species showed small quantity of gallic acid methyl ester by revealing faint violet-coloured spots, while the rest of the studied Acacia did not seem to contain these two compounds. Also catechin and naringenin were only detected in the two subspecies of A. nilotica.

3.3 Palynological Study

The microscopical examination of the pollen of the studied Acacia species showed that, 8 polyads were found in each anther with 4 polyads in each chamber. With the exception of A. laeta the polyads consisted of 16 monads with 8 monads centrally placed forming a sort of cubical block in which the individuals tended to be arranged in a rectangular fashion. The central group was surrounded by eight peripheral monads. According to the observed pollen characters two types of pollen were obtained, Type I pollen showed polyads with 78-102 µ in diameter and broken exine outline, included A. ehrenbergiana, A. nilotica, A. seyal, A. sieberana and A. tortilis (Figure 3-a).
whereas Type II pollen displayed polyads with 46-97 µm in diameter and continuous exine, included *A. mellifera* and *A. senegal* (Figure 3-b). *A. laeta* had polyad of 16 monads as well as 18-20 monads which were irregular and distorted (Figure 3-c).

4. Discussion

According to the TLC chromatographic patterns the studied species can be classified into three major groups:

Group I: includes species reacted positively to FeCl₃ reagent, with major violet-coloured spots. This group includes; *A. ehrenbergiana*, *A. nilotica* and *A. seyal*.

Group II: includes species reacted positively to AlCl₃ reagent with major yellow-coloured spots. This group includes; *A. mellifera*, *A. laeta* and *A. senegal*.

Group III: includes species exhibiting positive reactivity towards both FeCl₃ and AlCl₃ reagents revealing spots with violet and yellow colours respectively. This group includes; *A. sieberana* and *A. tortilis*.

The comparison of the species from the third group with Groups I and II showed that, they displayed the same major two violet-coloured spots as those of Group I but the colour intensity of one of them was lower in this Group. Moreover although they had the same degree of reactivity towards AlCl₃ as those from Group II, they differed considerably in the spots polarity (Figure 1).

TLC comparison of the isolated compounds with the flowers EtOAc extracts of the studied *Acacia* species suggested that gallic acid and its methyl ester were persistent in *A. ehrenbergiana*, *A. nilotica*, *A. seyal*, *A. sieberana*, and *A. tortilis* indicative of their strong chemical affinity. However, *A. sieberana* and *A. tortilis* were quite distinct from the others by their low concentration in gallic acid methyl ester as well as the presence of fluorescence spots that gave positive reaction with AlCl₃ reagent. Morphologically these two species were characterized by creamy-coloured flowers, while the rest of species were with orange-coloured flowers. Palynology all the above mentioned species had polyad of 16 monads and broken exine outline which is in accordance with the work of Guinet (1990).

*A. laeta*, *A. mellifera* and *A. senegal* were characterized by displaying fluorescent and AlCl₃ reactive spots. The chromatographical data showed that *A. laeta* exhibited stronger affinity towards *A. mellifera*. Palynologically, they all with the exception of *A. laeta*, had smaller size 16-celled polyads and continuous exine outline. *A. laeta* was different in having polyads both with the normal 16 monads and with 18-20 monads, which were slightly irregular and distorted. These characters supported the possibility of its hybrid origin from *A. mellifera* and *A. senegal* (Jackson & Peake, 1955; Brenan, 1955). Elamin (1972) suggested that the closer affinity of *A. laeta* to *A. mellifera* was probably due to back crossing with it in the past. The results obtained in the present study were generally supportive to Elamin’s classification. The relationships of the studied *Acacia* species are summarized in Table 1.
Table 1. Relationship of some morphological, palynological and phytochemical characters of the studied *Acacia* species

| Species          | Inflorescence type | Vegetative character | Palynological character | TLC behaviour | Elamin’s proposed group |
|------------------|--------------------|----------------------|-------------------------|---------------|-------------------------|
|                  |                    |                      |                         |               |                         |
| *A. ehrenbergiana* | Capitate           | Spinescent           | Spine                   | 16 Broken     | FeCl₃                   | Violet I                |
| *A. nilotica* ssp nilotica | Capitate           | Spinescent           | Spine                   | 16 Broken     | FeCl₃                   | Violet I                |
| *A. nilotica* ssp tomentosa | Capitate           | Spinescent           | Spine                   | 16 Broken     | FeCl₃                   | Violet I                |
| *A. seyal*       | Capitate           | Spinescent           | Spine                   | 16 Broken     | FeCl₃                   | Violet I                |
| *A. sieberana*   | Capitate           | Spinescent           | Spine                   | 16 Broken     | AlCl₃ & FeCl₃           | Yellow & violet I       |
| *A. tortilis*    | Capitate           | Spinescent           | Spine                   | 16 Broken     | AlCl₃ & FeCl₃           | Yellow & violet I       |
| *A. mellifera*   | Spicate            | Nonspinescent        | Prickles                | 16 Continuous | AlCl₃                   | Yellow II               |
| *A. laeta*       | Spicate            | Nonspinescent        | Prickles                | 16 and 16-20  | Continuous              | AlCl₃                   | Yellow II               |
| *A. senegal*     | Spicate            | Nonspinescent        | Prickles                | 16 Continuous | AlCl₃                   | Yellow II               |

In conclusion, the data obtained from both phytochemical and palynological studies were useful for taxonomic purposes in the Sudan *Acacia*. The preliminary phytochemical screening and isolation of some phenolic compounds pointed out the necessity of characterization of phenolic compounds present in all Sudan *Acacia* in order to obtain a better insight of the pattern of their occurrence in different taxonomical groups, as well as to evaluate their potentiality as chemotaxonomical markers.

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