Synthesis of Some New Functionalized Bis-thiazolidin-5-one and Bis-thiazolidin-4-one Derivatives

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Abstract A series of bis-thiazolidin-5-one and bis-thiazolidin-4-one derivatives was prepared by the treatment of highly functional thiocarbamoyl intermediates 2 and 4 with chloroacetyl chloride and chloroacetic acid, respectively. Treatment of the bis-thiocarbamoyl derivative 4a with diazotized sulphanilic acid affected acetyl cleavage to afford the corresponding arylhydrazono-thiocarbamoyl derivative 6. The title compounds bis-thiazolidin-5-one and bis-thiazolidin-4-one derivatives showed high reactivity towards azo coupling reaction with diazotized sulphanilic acid and Knoevenagel reaction with 4-isopropylbenzaldehyde.

Keywords Thiocarbamoyl, Chloroacetyl chloride, Bis-thiazolidin-5(and 4-)-ones, Japp-Klingmann Reaction

1. Introduction

The chemistry of thiazole derivatives, including new methodologies for their preparation, and recent applications, such as their growing use in organic synthesis in the biological field and asymmetric catalysis as ligands has been recently reviewed[1]. Thiazolidinones are known mainly as biologically active compounds with a broad range of activity and as intermediates in the synthesis of antibiotics and dyes[2-4]. Several papers have been published on the use of these compounds as antimicrobial[5,6], antifungal[7,8], anti-inflammatory activity[9], anticonvulsant[10], anesthetic [11] and antiviral drugs[12]. 2-Aminothiazoles and its derivatives are also used in the synthesis of various types of dyes[13-16]. In this work, a series of bis-thiazolidin-5-one and bis-thiazolidin-4-one derivatives was prepared by treatment of their corresponding thiocarbamoyl intermediates with chloroacetyl chloride and chloroacetic acid, respectively. Also, the reactions of title compounds bis-thiazolidin-one derivatives with diazotized sulphanilic acid and with 4-isopropylbenzaldehyde were studied.

2. Materials and Methods

All melting points were measured on an electrothermal Gallenkamp melting apparatus. Elemental analyses were carried out at the Microanalytical Unit, Faculty of Science, University of Mansoura, Egypt; the results were in satisfactory agreement with the calculated values. IR spectra (KBr) were determined on a Mattson 5000 FTIR spectrometer (not all frequencies are reported). Mass spectra were obtained at a Finnigan MAT 212 instrument (electron impact: 70 eV).

The title compound 1 was prepared by boiling a hot solution of ethyl acetoacetate (7.6 ml, 0.06 mol) in 30 ml dry xylene with a hot solution of benzidine (5.5 g, 0.03 mol) in dry xylene (30 ml). The precipitate that formed on cooling was filtered off, dried and recrystallized from ethanol to give compound 1 as buff crystals, m.p. = 230 - 232 °C, yield = 85%. IR (ν/cm⁻¹): 3356 (NH), 1668 (C=O), 1633 (C=O). MS: m/z = 3356 (NH), 1668 (C=O), 1633 (C=O), Elemental analysis: Calc. for C₂₀H₂₀O₄N₂ (352.38): C = 68.16, H = 5.72, N = 7.95%. Found: C = 68.29, H = 5.81, N = 7.88%.

Synthesis of N,N’-[1,1’-biphenyl]-4,4’-diy]-bis-(3-oxo-2-(5-oxo-3-phenyl-thiazolidin-2-ylidene)butanamide) (3)

To a cold suspension of finely divided KOH (0.78 g, 14 mmol) in DMF (20 ml) was added the title compound 1 (2.46 g, 7 mmol) followed by phenyl isothiocyanate (1.68 ml, 14 mmol). The mixture was stirred at room temperature overnight, and then treated with chloroacetyl chloride (1.12 ml, 14 mmol). The stirring was continued at room temperature for 8 hours. The reaction mixture was poured into ice-cold water. The resultant solid product was collected by filtration, washed with water, dried, and recrystallized from ethanol. m.p. = 223 - 225°C, yield = 72%. IR (ν/cm⁻¹): 3356 (NH), 1730 (C=O), 1660 (C=O), 1633 (C=O), MS: m/z = 702 (M⁺, 20), 400 (30), 298 (45), 226 (83), 103 (75), Elemental analysis: Calc. for C₃₈H₃₈N₄O₆S₂ (702.8): C =...
The diazotized solution of sulphamic acid (0.69 g, 2 mmol) was added with continuous stirring to a cold solution of compound 3 or compound 5a (4 mmol) in 20 ml ethanol containing sodium acetate (0.75 g). The reaction mixture was stirred at 0-5°C for 2 hours, left to stand at room temperature. The solid product that obtained was filtered off, dried and recrystallized from ethanol to afford compound 7 or compound 8, respectively.

(7): m.p. = 235 – 237°C, yield = 73%. IR (v/cm⁻¹): 3430 (NH), 1688 (C=O), 1657 (C=O). MS: m/z = 1071 (M⁺, 15), 974 (18), 408 (25), 271 (33), 120 (63), 58 (100), 53 (66). Elemental analysis: Calc. for C₃₈H₃₉N₈O₆S₂ (963.17): C = 72.33, H = 5.23, N = 5.82%. Found: C = 72.24, H = 5.18, N = 5.70%.

(8): m.p. > 320°C, yield = 76%. IR (v/cm⁻¹): 3395 (NH), 1673 (C=O), 1633 (C=O). MS: m/z = 1071 (M⁺, 5), 980 (23), 311 (33), 135 (55), 79 (100), 65 (73). Elemental analysis: Calc. for C₃₈H₃₉N₈O₆S₂ (1071.14): C = 56.06, H = 3.58, N = 10.46%. Found: C = 56.32, H = 3.62, N = 10.24%.

Formation of compounds 9 and 10

A mixture of bis-thiazolidin-5-one derivative 3 or bis-thiazolidin-4-one derivative 5a (3 mmol) and 4-isopropylbenzaldehyde (0.9 ml, 6 mmol) was refluxed for 4 hours in ethanol (30 ml) containing catalytic amount of piperidine. The solid products that formed on cooling were filtered off, dried and recrystallized from ethanol to afford the corresponding arylidene derivatives 9 or 10, respectively.

(9): m.p. = 310 – 312°C, yield = 78%. IR (v/cm⁻¹): 3362 (NH), 1678 (C=O), 1654 (C=O). MS: m/z = 963 (M⁺, 30), 893 (45), 507 (35), 210 (40), 184 (100). Elemental analysis: Calc. for C₅₈H₅₀N₈O₁₂S₄ (1073.17): C = 72.33, H = 5.23, N = 5.82%. Found: C = 72.24, H = 5.18, N = 5.88%.

(10): m.p. > 350°C, yield = 73%. IR (v/cm⁻¹): 3380 (NH), 1666 (C=O), 1643 (C=O). MS: m/z = 963 (M⁺, 15), 601 (20), 386 (100), 236 (77), 92 (95), 51 (82). Elemental analysis: Calc. for C₅₈H₅₀N₈O₁₂S₄ (963.17): C = 72.33, H = 5.23, N = 5.82%. Found: C = 72.24, H = 5.18, N = 5.88%.

3. Results and Discussions

The highly versatile compound N,N’-[([1,1’-biphenyl]-4,4’-diyl)]bis-(3-oxo-thiabicyclo[4.3.0]non-5-one) (I) was prepared from benzidine and ethyl acetate. The base-promoted reaction of the acidic methylene compound 1 with phenyl isothiocyanate in dry DMF at room temperature in basic medium led to the formation of the non-isolable intermediates 2 which underwent intramolecular cyclization reaction with chloroacetyl chloride to afford the corresponding bis-thiazolidin-5-one derivative 3 (Scheme 1).

In situ treatment of the non-isolable intermediates 2 with dilute HCl gave the corresponding bis-thiocarbamoyl derivatives 4a and 4b. Refluxing of 4 with chloroacetic acid in acetic acid containing anhydrous sodium acetate afforded the corresponding bis-thiazolidin-4-one derivatives 5. The molecular structure of 5 was confirmed by analytical and spectral data.
Treatment of bis-thiocarbamoyl derivative 4a with diazotized sulphanilic acid in the presence of sodium acetate affected acetyl cleavage (Japp-Klingemann reaction type) with the formation of arylhydrazono-thiocarbamoyl derivative 6 (Scheme 2). The structure of the highly functionalized arylhydrazono-thiocarbamoyl derivative 6 was elucidated on the basis of its elemental analysis and spectral data.

The reactivity of the methylene group in bis-thiazolidin-5-one derivative 3 and bis-thiazolidin-4-one derivative 5a was tested toward the azo coupling reaction with diazonium salts. Thus, when two moles of diazotized sulphanilic acid at 0-5 °C reacted with the thiazolidinonedervatives 3 and 5a yielded the corresponding bis-hydrazono derivatives 7 and 8 respectively. The structures of compounds 7 and 8 were assigned on the basis of their elemental analyses and spectral data.

Treatment of bis-thiazolidin-5-one derivative 3 and bis-thiazolidin-4-one derivative 5a with p-isopropylbenzaldehyde in ethanol/piperidine afforded the corresponding bis-(4-arylidene-thiazolidin-5-one) 9 and bis-(5-arylidene-thiazolidin-4-one) 10 in good yields. Assignment of the products 9 and 10 was based on their elemental analyses and spectral data.
Scheme 2. Synthesis of bis-thiazolidin-4-one derivatives 5 and arylhydrazono-thiocarbamoyl derivative 6

\[ \text{Ar} = p'\text{HO}_3\text{SC}_6\text{H}_5 \]

\[ \begin{align*}
\text{2} + \text{dil. HCl} & \rightarrow \text{4} \\
\text{a: R} = \text{C}_6\text{H}_5, \text{ b: R} = \text{C}_2\text{H}_5 \\
\text{ArN}_2\text{Cl} & \text{NaOAc/EtOH} \\
\text{CICH}_2\text{COOH} & \text{NaOAc/AcOH} \\
\text{5} & \\
\text{a: R} = \text{C}_6\text{H}_5, \text{ b: R} = \text{C}_2\text{H}_5
\end{align*} \]

Scheme 3. Reactions of thiazolidin-5-one derivatives 3 and 5a with diazotized sulphamic acid and \( p' \)-isopropylbenzaldehyde

\[
\begin{align*}
\text{7: Ar} &= p'\text{HO}_3\text{S-C}_6\text{H}_4 \\
\text{8: Ar} &= p'\text{HO}_3\text{S-C}_6\text{H}_4 \\
\text{9: Ar} &= p'(\text{CH}_3)_2\text{CH-C}_6\text{H}_4 \\
\text{10: Ar} &= p'(\text{CH}_3)_2\text{CH-C}_6\text{H}_4
\end{align*}
\]
4. Conclusions

We have synthesized new bis-thiazolidin-5-one and bis-thiazolidin-4-one derivatives by the reaction of available thiocarbamoyl derivatives with chloroacetyl chloride and chloroacetic acid, respectively. The title compounds bis-thiazolidin-5-one and bis-thiazolidin-4-one derivatives showed high reactivity towards azo coupling reaction with diazotized sulphanilic acid and Knoevenagel reaction with 4-isopropylbenzaldehyde.

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