Modeling of pyruvate decarboxylases from ethanol producing bacteria

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Abstract:
Pyruvate decarboxylase (PDC) is a key enzyme in homoethanol fermentation process, which decarboxylates 2-keto acid pyruvate into acetaldehyde and carbon dioxide. PDC enzymes from potential ethanol-producing bacteria such as Zymomonas mobilis, Zymobacter palmae and Sarcina ventriculi have different Km and kcat values for the substrate pyruvate at their respective optimum pH. In this study, the putative three-dimensional structures of PDC dimer of Z. palmae PDC and S. ventriculi PDC were generated based on the X-ray crystal structures of Z. mobilis PDC, Saccharomyces cerevisiae PDC form-A and Enterobacter cloacae indolepyruvate decarboxylase in order to compare the quaternary structures of these bacterial PDCs with respect to enzyme-substrate interactions, and subunit-subunit interfaces that might be related to the different biochemical characteristics. The PROCHECK scores for both models were within recommended intervals. The generated models are similar to the X-ray crystal structure of Z. mobilis PDC in terms of binding modes of the cofactor, the position of Mg2+, and the amino acids that form the active sites. However, subunit-subunit interface analysis showed lower H-bonding in both models compared with X-ray crystal structure of Z. mobilis PDC, suggesting a smaller interface area and the possibility of conformational change upon substrate binding in both models. Both models have predicted lower affinity towards branched and aromatic 2-keto acids, which correlated with the molecular volumes of the ligands. The models shed valuable information necessary for further improvement of PDC enzymes for industrial production of ethanol and other products.

Keywords: bacterial pyruvate decarboxylase; homology modelling; molecular docking, 2- keto acids.

Background:
The pyruvate decarboxylase enzyme (PDC, EC 4.1.1.1) can be found in plants and fungi, but is very rare in prokaryotes and absent in animals [1]. This enzyme requires thiamine diphosphate (ThDP) and Mg2+ cofactors for its function in catalyzing the non-oxidative decarboxylation of 2-keto-acids, of which its preferred substrate is pyruvate. Catalytically active ThDP, called ylide-ThDP is formed during decarboxylation of 2-keto-acids, of which its preferred substrate is pyruvate [2]. The acetaldehyde produced by PDC pyruvate decarboxylation is then converted to ethanol by alcohol dehydrogenase (ADH, EC 1.1.1.1). In addition, PDC decarboxylates 2-keto acid pyruvate into acetaldehyde and carbon dioxide. PDC enzymes from potential ethanol-producing bacteria such as Z. mobilis, Z. palmae and S. cerevisae possess different Km and kcat values for the substrate pyruvate at their respective optimum pH. In this study, the putative three-dimensional structures of PDC dimer of Z. palmae PDC and S. ventriculi PDC were generated based on the X-ray crystal structures of Z. mobilis PDC, Saccharomyces cerevisiae PDC form-A and Enterobacter cloacae indolepyruvate decarboxylase in order to compare the quaternary structures of these bacterial PDCs with respect to enzyme-substrate interactions, and subunit-subunit interfaces that might be related to the different biochemical characteristics. The generated models are similar to the X-ray crystal structure of Z. mobilis PDC in terms of binding modes of the cofactor, the position of Mg2+, and the amino acids that form the active sites. However, subunit-subunit interface analysis showed lower H-bonding in both models compared with X-ray crystal structure of Z. mobilis PDC, suggesting a smaller interface area and the possibility of conformational change upon substrate binding in both models. Both models have predicted lower affinity towards branched and aromatic 2-keto acids, which correlated with the molecular volumes of the ligands. The models shed valuable information necessary for further improvement of PDC enzymes for industrial production of ethanol and other products.

Calculation of ligand volume
The volume of the ligand was calculated using Mol Volume version 1.18. The van der Waals radii for each atom type was derived from the CHARMM 22 force field. The radius of the spherical probe (R_PROBE) was set to 2.0 Å, and the GRID_STEP was set to 0.5 Å.

Protein-ligand docking
The AutoDockTools version 1.4.6 [19] was used to prepare the ligand and the protein structures, and to generate the grid parameter and docking parameter files. The 2-keto acid structures were generated using ChemDraw Ultra 7.0 [20]. The ThDP and Mg2+ were included in docking calculations. The whole protein was treated as a rigid body while the ligand was allowed to move freely.
in the grid box during the docking process. The grid box was generated using AutoDock version 4 [21]. The size of the grid box was set to 40 x 40 x 40 points on each side with the distance of 0.375 Å between each point.

Docking simulations were carried out using the AutoDock version 4 [21] with the Lamarckian genetic algorithm (LGA). Each docking calculation was derived from 100 different runs that were set to terminate after a maximum of 2 500 000 energy evaluations. The point size was set to 150. The other parameters were set to the default value provided in the AutoDock4. AutoDock4 calculates Inhibition constant, K as “exp [(ΔG X 1000) / (Rcal X TK)]” where ΔG is docking energy which is a sum of intermolecular energy and internal energy of ligand, Rcal = 1.98719 cal K mol⁻¹ and TK = 298.15 K. The equilibrium dissociation binding constant, Kd is calculated as “1/K,” that is commonly used to describe the affinity between a ligand and a protein.

Discussion:
Homology models of ZpPDC and SvPDC
In general, the homology models of ZpPDC and SvPDC are similar to the X-ray crystal structure templates. The root mean square deviation (RMSD) between the Cu of ZpPDC and ZmPDC, and between the Cu of SvPDC and E. coliPDC were 2.27 and 2.93 Å respectively (Figure 2). For ZpPDC, 90.4% of the residues were in most favorable regions; only two residues (Ser73 from each subunit) were in disallowed regions giving a PROCHECK score of -0.32. For SvPDC, 88.7% of the residues are in most favorable regions with only four residues (Arg187, Arg494, and Thr520 of subunit A and Lys333 of subunit B) in disallowed regions, with a PROCHECK score of -0.33. However, these residues were not involved in cofactor or substrate binding according to the equivalent residues in other species. After a short energy minimization, these residues were in the generously allowed regions.

Each subunit of both models consists of three domains, namely PYR, R and PP domains. All share the same α/β topology common to all thiamine dependent enzymes. The PYR and PP domains have six parallel beta sheets flanked by six alpha helices, whereas the R domain contains four parallel beta sheets (β4-7) with a pair of adjacent anti-parallel beta sheet (β2-3) comparable to ZmPDC [5] (Figure 1). In all PDCs, two PDC monomers interact to form a dimer which plays a vital role in the catalytic mechanism of the enzyme since it comprises the active site. In both models, most of the interface residues from the PYR and PP domains come from R domain. The contribution of interface residues by each domain reflects the average mobility of the domain in the resolved PDC crystal structure: e.g. for E. coliPDC the PYR domain has the lowest average B-factor, 23 Å² whereas the R domain has the highest overall B factor, 37 Å² [8]. The ZpPDC interface accessible surface area is similar to ZmPDC (19%), but different from SvPDC (16%). Fewer H-bonds are present in the SvPDC model (30) compared with the ZpPDC model (44) and ZmPDC (66); in contrast the E. coliPDC has the fewest (26). The number of hydrogen bonds at the interface between subunits was found to be proportional to the subunit interface area for experimentally determined structures with correlation coefficient of 0.75 (Figure 3). Extrapolation of this relationship was used to determine interface areas for ZpPDC and SvPDC, which were 3587 Å² and 3264 Å² respectively. These interfaces are much smaller than the ZnPDC interface area, and are comparable with ScPDC form B and E. coliPDC, suggesting some conformational changes in ZpPDC and SvPDC upon substrate binding.

Binding modes of the cofactors
In our homology models, the pyrimidine ring of ThDP interacted with the PYR domain of one subunit, and the rest of the molecule interacted with the PP domain of the other subunit giving “V” conformation of ThDP [22] that is in agreement with resolved X-ray crystal structures of PDCs [5, 16]. Only few acidic and basic residues that were in close contact with the cofactors among the X-ray crystal structures and the models were similar i.e. Glu50, Asp27, His113, Glu473 (Table 1 in supplementary material).

Substrate binding sites of the homology models
In the ZmPDC X-ray crystal structure, the substrate atoms are surrounded by the side chains of Asp26, His113, His114, Tyr290, Thr388 and Glu473 (all conserved except Tyr290) which might be involved in substrate binding and catalysis [5]. Superimposition of three X-ray crystal structures (ScPDC form-B [16], ZmPDC and E. coliPDC) and the two homology models of ZpPDC and SvPDC showed equivalent conserved residues were also present in homology models with similar orientation, while Tyr290 of ZmPDC and ZpPDC was replaced by smaller residue, Thr288 in SvPDC and E. coliPDC (Figure 4). Conversion of Thr288 to its ZmPDC analogue would decrease active site volume of SvPDC and subsequently decrease Kd value for pyruvate. The two conserved histidine residues (His113, His114 for ZmPDC and SvPDC; His112, His113 for ZpPDC) that have crucial roles in catalysis [23] and are in very close proximity to the pyrimidine ring of ThDP are orientated similar to X-ray crystal structure of ZmPDC (Figure 4).

Pyruvamide interacts with ThDP and side chains of Asp26, His113, Glu468 of ZpPDC, and Asp27, His114 of SvPDC. The distances between pyruvamide and these structures are within 4 Å (Figure 5A and 5B). The orientation of bulky residues in the active site is similar in ZmPDC and ZpPDC (Tyr392 and Thr387, respectively) (Figure 4) while Tyr384 is found in SvPDC model at equivalent position with different orientation. This residue is replaced by alanine in ScPDC (Ala392) and E. coliPDC (Ala387). The type and position of these residues indicate that ZpPDC and SvPDC may have similar carboligation activity as of ZmPDC, since the second aldehyde molecule in the carboligation reaction is accommodated in this region [5].

Substrate activation of homology models
Pyruvamide has been shown to display allosteric activation by the substrate with a Kd value of 4 to 6 mM [7], similar to the fungal and plant PDCs, while ZmPDC, ZpPDC [6] and E. coliPDC [8] follow Michaelis-Menten kinetics. Superimposition of the ScPDC form-B with the homology model of ScPDC revealed Tyr155, Tyr321, Glu158, Ala221, Arg220, Lys225, within 5 Å distance of pyruvamide. Hydrogen bonds are formed between the amide group of pyruvamide and the side chain hydroxyl oxygen atom of Tyr155 and the main chain oxygen atom of Arg220 of same subunit (Figure 5C and 5D), which are similar as in ScPDC form-B. Comparison of X-ray crystal structures of non-activated ScPDC form-A and activated ScPDC form-B show significant changes in the interfaces: between the R and the PYR domain, between monomer subunits, and between dimer subunits, suggesting the interaction of pyruvamide with ScPDC leads to large scale conformational changes during catalysis [14]. The SvPDC model suggests that similar conformational changes upon substrate binding at regulatory site could occur in this model. Another theory of activation by covalent binding of substrate to Cys221 which then interacts with His92 of ScPDC form-B [16] seems implausible for SvPDC since these residues are replaced by Glu216 and Asn91 at the equivalent positions respectively.

Affinity of ZpPDC and SvPDC towards other 2-keto acids
PDC homologues across species have conserved cofactor binding residues but different substrate preferences. The experimentally-determined Kd values for four different 2-keto acids binding to ZmPDC [3] suggest a relationship between molecular volume of substrate and Kd. (Figure 6); therefore, dissociation values (Kd) for various 2-keto acids of varying molecular volumes docked with crystal structure of ZmPDC, ZpPDC and SvPDC models were determined. The larger 2-keto acids bind with higher Kd than those of the smaller 2-keto acids to both homology models (Table 2 in supplementary material), suggesting that the substrate binding preferences of the ZpPDC and SvPDC enzymes are similar to ZmPDC. Ile472 and Thr382 residues proposed to be responsible for making substrate binding site volume small in ZmPDC [8] are conserved in SvPDC (equivalent residues Ile464, Thr382) and ZpPDC (equivalent residues Ile467, Thr380). While in E. coliPDC and
Hypothesis

benzoylformate dehydrogenase whose preferred substrates are aromatic 2-keto acids, the equivalent residues at same positions are Val467, Gln383 and Ala460, Leu403, respectively. Therefore, we speculate that mutation of Ile467 of ZpPDC and Ile464 of SvPDC to small residues like alanine or valine might produce variants capable to use aromatic 2-keto acids as substrate.

Figure 1: Multiple amino acid sequence alignment of ZmPDC, ZpPDC, SvPDC, EcIPDC and ScPDC (1pvm). Identical, functionally conserved and semi-conserved amino acid residues are shaded. Dashes indicate gaps introduced in alignment. The ThDP-binding motif is underlined with broken lines. Secondary structural motifs are indicated above the alignment according to the X-ray structure of ZmPDC [5]. (▲) are residues within 3 Å of ThDP in ZmPDC.

Figure 2: Superimposition of the Co of the X-ray crystal structure of ZmPDC (Green) and the homology model of ZpPDC (Red) (A), and the X-ray crystal structure of EcIPDC (Yellow) and the homology model of SvPDC (Blue) (B). The ThDP and Mg$^{2+}$ are shown in space-filled model in magenta and dark blue, respectively.
Figure 3: Correlation of interface area and number of hydrogen bonds of crystal structures of ZmPDC (1zpd) [5]; Pyruvate oxidase from Lactobacillus plantarum (1pox) [24]; EcPDC (1ovm) [8]; ScPDC form-B (1qpb) [16]; ScPDC form-A (1pvd) [9] and phenylpyruvate decarboxylase of Azospirillum brasilense (2nxw) [25].

Figure 4: Superimposition of crystal structures of ZmPDC (Red), EcPDC (Yellow), ScPDC form-A (Pink), and homology models of ZpPDC (Blue) and SvPDC (Green) showing different orientation of first Histidine residue of two consecutive Histidine residues. Residues are numbered according to ZmPDC. Residues labeled with (*) are from the other monomer. Figure generated using DeepView / Swiss-PdbViewer 3.7.
**Figure 5:** Putative active sites of homology models of ZpPDC (A) and SvPDC (B). Asp27* in figure B is shown in line structure for clarity. The regulatory site of SvPDC is shown in (C) and in close detail in (D). Cofactors, ThDP (Green) and Mg$^{2+}$ (Blue), pyruvamide (PYM) (Red) at the regulatory site of SvPDC are shown in space-filled model in figure C. Residues labeled with (*) are from the other monomer. Potential H-bonds are indicated as red dotted lines.

**Figure 6:** Correlation of ZmPDC experimental $K_m$ values for four 2-keto acids; pyruvate, 2-ketobutanoic acid, 2-ketopentanoic acid, and 2-ketohexanoic acid, [23] and their calculated molecular volume.

**Conclusion:**
In this study, the homology models of the pyruvate decarboxylase of *Z. palmae* and *S. ventriculi* were generated to explore the structural similarities and differences among bacterial PDCs at the atomic level. The homology models and the X-ray crystal structure of ZmPDC were similar in terms of cofactor binding modes,
substrate binding residues, and active site volume. The mechanism of allosteric activation shown by SvPDC could be similar as in ScPDC form-B which needs further experimental verification. A relationship was found between the molecular volume of 2-ketoacid substrates and calculated dissociation constant $K_d$ values, demonstrating preference of ZpPDC and SvPDC enzymes for aliphatic 2-ketoacids.

Authors' contributions:
AS performed modelling, docking and wrote the first draft of the manuscript. SD helped with data analysis and provided suggestion on manuscript. EJ supervised the project, provided mentorship and helped with drafting and revising of the manuscript. All authors read and approved the final manuscript.

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### Table 1: Residues within 3 Å of ThDP in different X-ray crystallography structures and homology models.

| Protein (PDB ID)     | Contacted residues                                    |
|----------------------|-------------------------------------------------------|
| ScPDC (1pvd)         | Glu51*, Thr390, Gly413, Asp444, Gly445, Ser446, Thr475, Ile476 |
| EcIPDC (1ovm)        | Glu52*, Thr385, Gly408, Ala437, Gly464, Thr466       |
| ZmPDC (1zpd)         | Glu50*, Gly413, Asp440, Gly441, Ser442, Asn467, Tyr470, Thr471, Ile472 |
| ZpPDC (model)        | Ala24*, Glu49*, Gly408, His409, Asp435, Gly436, Ser437, Asn462, Gly464, Tyr465, Val466, Ile467 |
| SvPDC (model)        | Glu50*, Ser434, Met405, Ile407, Gly461, Thr462, Thr463, Ile464 |

Residues labeled with (*) are from the other monomer.

### Table 2: Calculated $K_i$ and $K_d$ of 2-keto acid derivatives against the X-ray crystal structure of ZmPDC and homology models of ZpPDC and SvPDC.

| Ligands                     | ZmPDC | ZpPDC | SvPDC |
|-----------------------------|-------|-------|-------|
|                            | $K_i$ (mM) | $K_d$ (mM) | $(1/K_d)$ (mM) | $K_i$ (mM) | $K_d$ (mM) | $(1/K_d)$ (mM) | $K_i$ (mM) | $K_d$ (mM) | $(1/K_d)$ (mM) |
| Pyruvate                    | 497   | 371   | 0.003 | 692   | 0.001 | 920 | 0.001 |
| 2-ketobutanoic acid         | 568   | 719   | 0.001 | 534   | 0.002 | 583 | 0.002 |
| 2-ketopentanoic acid        | 634   | 276   | 0.004 | 330   | 0.003 | 309 | 0.003 |
| 2-ketohexanoic acid         | 706   | NA    | NA    | 166   | 0.006 | 185 | 0.005 |
| 2-keto-3-methylbutanoic acid| 597   | 662   | 0.002 | 267   | 0.004 | 310 | 0.003 |
| 2-keto-3-methylpentanoic acid| 845  | 717   | 0.001 | 148   | 0.007 | 147 | 0.007 |
| 2-keto-3-cyclohexylpropanoic acid| 825 | NA    | NA    | NA    | NA | NA | NA |
| 2-keto-3-3-dimethylpentanoic acid| 735 | NA    | NA    | 80    | 0.012 | 75  | 0.013 |
| 2-keto-4-methylhexanoic acid| 756   | 786   | 0.001 | 73    | 0.014 | 98  | 0.010 |
| 2-keto-4-methylpentanoic acid| 694  | 580   | 0.002 | 130   | 0.008 | 140 | 0.007 |
| 2-keto-4,4-dimethylpentanoic acid| 700 | NA    | NA    | 111   | 0.009 | 99  | 0.010 |
| Acetaldehyde                | 373   | 41    | 0.024 | 44    | 0.022 | 27  | 0.038 |

NA: Autodock calculated free energy of binding was positive.