Ultra-high open-circuit voltage of tin perovskite solar cells via an electron transporting layer design

Xianyuan Jiang¹,³, Fei Wang¹,³, Qi Wei¹, Hansheng Li¹, Yuequn Shang¹, Wenjia Zhou¹, Cheng Wang¹,², Peihong Cheng¹, Qi Chen², Liwei Chen² & Zhijun Ning¹

Tin perovskite is rising as a promising candidate to address the toxicity and theoretical efficiency limitation of lead perovskite. However, the voltage and efficiency of tin perovskite solar cells are much lower than lead counterparts. Herein, indene-C₆₀ bisadduct with higher energy level is utilized as an electron transporting material for tin perovskite solar cells. It suppresses carrier concentration increase caused by remote doping, which significantly reduces interface carriers recombination. Moreover, indene-C₆₀ bisadduct increases the maximum attainable photovoltage of the device. As a result, the use of indene-C₆₀ bisadduct brings unprecedentedly high voltage of 0.94 V, which is over 50% higher than that of 0.6 V for device based on [6,6]-phenyl-C₆₁-butyric acid methyl ester. The device shows a record power conversion efficiency of 12.4% reproduced in an accredited independent photovoltaic testing lab.
Halide perovskites are rising as star materials for next-generation solar cells. However, the state of the art lead perovskite still needs to be upgraded since: firstly, the heavy metal character of lead may cause an environmental concern; secondly, according to Shockley–Queisser limitation, the highest efficiency of lead perovskite is <31% due to its large bandgap (1.55 eV). The development of alternative perovskite materials is therefore highly desirable. Tin perovskite is regarded as an ideal candidate due to its narrow bandgap and environmental benign character. Up to now, tin perovskite solar cell (PSC) demonstrates the highest efficiency among all kinds of lead-free perovskites.

Despite the rapid development of tin PSCs, the efficiency is much lower than lead perovskite nowadays. One critical factor is the extremely low open-circuit voltage. High defect density in the device from 0.3 V to 0.6 V (ref. 13). However, compared to its counterpart, e.g., addition of reductive additives demonstrates the highest efficiency for over 12,4%, which is almost 30% higher than the highest efficiency reported up to now.

We performed spectroscopy characterization to study the structure of the films further. The absorption edge of the film based on PEA15-SCN (905 nm) is quite close to that of three-dimensional (3D) structure PEAO (917 nm; Supplementary Fig. 7a), implying the presence of 3D-like structures in the film. In combination with GIWAXS measurement, it can be speculated that part of PEA molecules are consumed for constructing two-dimensional (2D) and 2D-like structures, which are mixed with 3D structures in the film (Supplementary Fig. 8). The absorption spectrum of PEA15-SCN demonstrates a lower Urbach energy of ~65 meV (Fig. 1c), which indicates the reduced defect concentration with the addition of NH4SCN (refs. 25,26). Scanning electron microscope (SEM) images of PEA15 and PEA15-SCN films are similar and both of them show smooth morphology (Fig. 1d).

**Device structure design.** We measured the energy levels of the perovskite films for device structure design. Based on ultraviolet photoelectron spectroscopy (UPS), the Fermi level and valence band minimum for PEA15-SCN perovskite film (Fig. 2a) are calculated to be –4.52 eV and ~5.08 eV, respectively. Combining the optical bandgaps of 1.39 eV for PEA15-SCN determined by Tauc plots (Supplementary Fig. 7b), the conduction band maximum of perovskite film is calculated to be ~3.69 eV for PEA15-SCN (Fig. 2b). Considering the shallow conduction band position of tin perovskite, the typically used ETLS such as PCBM shows a large energy level offset due to its much deeper band position. Since the maximum attainable photovoltage is determined by the quasi-Fermi level splitting, the use of shallower energy level ETLS could increase the Voc of the device. Therefore, we used ICBA as ETLS, since it has a LUMO level of ~3.74 eV shallower than the commonly used PCBM of ~3.91 eV.

**Device performance.** We fabricated the device based on PEA15-SCN films above using ICBA and PCBM as ETLS, respectively. The device based on ICBA achieved a much-enhanced voltage up to 0.94 V (Fig. 3a). The efficiency certified in an independent lab is up to 12.4% (Voc = 0.94 V, Jsc = 17.4 mA cm–2, FF = 75%; Table 1, Supplementary Fig. 9). This is ~30% higher than the best value reported up to now20–22. In contrast, the device based on PCBM shows much worse efficiency of 7.7% (Fig. 3b, Supplementary Fig. 10), and the voltage is only 0.6 V. The device based on ICBA shows a low hysteresis as the current density–voltage (J–V) curves under reverse and forward scan overlapped well (Fig. 3b). The integrated photocurrent density (17.3 mA cm–2) obtained from the external quantum efficiency (EQE) spectra of PEA15-SCN device (Fig. 3c) agrees closely with current density (Jsc) in the J–V curves. Moreover, the device performance shows good reproducibility (Fig. 3d, Supplementary Fig. 11).

We then fabricated devices based on PEA15 and ICBA for comparison, which shows Voc of 0.78 V, much smaller than that of the film with PCBM (Table 1). Despite the current density of the film is comparable to that based on PEA15-SCN, the overall efficiency of the device is only 10.1%.

**Perovskite film structure.** Perovskite films are prepared using a typical anti-solvent method via a one-step process (details can be found in Methods section). We tune the concentration of PEA (PEA = C24H48CH2CH2NH3+) in precursors to get the PEAxSnI3–SnI3 (FA = HC(NH2)2) perovskite (abbreviated as PEAx, x = 0 and 15). Films are prepared on poly(3, 4-ethylenedioxythiophene):poly(styrene sulfonate) (PEDOT) substrates to control the nucleation and orientation of perovskite. NH4SCN (SCN) is used as additive to modify perovskite growth. NH4SCN (SCN) is used as additive to modify perovskite growth. We characterized the crystal structure by X-ray diffraction (XRD). The diffraction peaks of the perovskite film with NH4SCN (PEA15–SCN) show a higher intensity (Fig. 1a) and a smaller full-width half-maximum (FWHM) value (Fig. 1b), indicating a larger domain size for PEA15–SCN film.

We then studied the microstructure of perovskite film by grazing-incidence wide-angle X-ray scattering (GIWAXS). The strong diffraction spot in 90 degree shows that PEA15–SCN film is perpendicularly grown on the substrate along [100] direction. Both films with and without SCN grown on PEDOT show similar diffraction spots from single layer and double layer structures at both small and large grazing-incidence angles (Supplementary Fig. 3). This is completely different from the image of hierarchy structure grown on NiOx substrate. We conclude that low dimensional structures homogeneously distribute in the film, which can be ascribed to the higher binding energy of PEA and FA molecules on PEDOT substrate, as calculated by density functional theory (DFT) simulation (Supplementary Fig. 6, Supplementary Table 1).

We performed spectroscopy characterization to study the structure of the films further. The absorption edge of the film based on PEA15–SCN (905 nm) is quite close to that of three-dimensional (3D) structure PEAO (917 nm; Supplementary Fig. 7a), implying the presence of 3D-like structures in the film. In combination with GIWAXS measurement, it can be speculated that part of PEA molecules are consumed for constructing two-dimensional (2D) and 2D-like structures, which are mixed with 3D structures in the film (Supplementary Fig. 8). The absorption spectrum of PEA15–SCN demonstrates a lower Urbach energy of ~65 meV (Fig. 1c), which indicates the reduced defect concentration with the addition of NH4SCN (refs. 25,26). Scanning electron microscope (SEM) images of PEA15 and PEA15–SCN films are similar and both of them show smooth morphology (Fig. 1d).
The PEA15-SCN device shows good shelf stability: the encapsulated device maintained 90% of the initial performance for as long as 3800 hours (Fig. 3e). Steady-state power conversion efficiency (PCE) measurement was carried out to evaluate the operation stability of the device. The device showed stable efficiency at continuous operation at maximum power point for 150 s (Fig. 3f). However, under longer time illumination, the efficiency decay slows, and the device loses 50% of its initial performance after continuous operation for over 300 min under amplitude modulation (AM) 1.5 illumination (Supplementary Fig. 13). The efficiency decay might be ascribed to the ion migration or accumulation of carriers at the interface.

**Device characterization.** To clarify the mechanism for the much-enhanced voltage and performance of the device based on ICBA, we studied the interface recombination between perovskite and ETL. Firstly, scanning Kelvin probe microscopy (SKPM) was used to measure the contact potential difference between the perovskite layer and the top contact. The results showed a lower contact potential difference for the PEA15-SCN device compared to the PEA15 device, indicating a lower recombination rate at the interface. This is consistent with the improved performance of the PEA15-SCN device.

![Figure 1: Perovskite film characterization. a XRD spectra of perovskite films and b the corresponding FWHM values of (100) peak. c The Urbach energy of perovskite films. d SEM images of perovskite films. The scale bar is 1 μm.](image1)

![Figure 2: Band structure of tin PSCs. a UPS spectra of secondary electron cutoff and valence band of perovskite films. b Schematic illustration of energy levels. Dashed lines represent the quasi-Fermi level of ICBA (EFn-ICBA), PCBM (EFn-PCBM), and PEDOT (EFp).](image2)
employed to analyze interfacial energy level alignment of perovskite/ETL interface (Fig. 4a, Supplementary Figs. 14 and 15)\textsuperscript{29}. The surface potential of ICBA is 20 mV higher than perovskite, indicating that the Fermi level of ICBA is 20 mV higher than that of perovskite. Similarly, we deduced that the Fermi level of PCBM is ~100 mV higher than perovskite (Fig. 4a). Hence, it can be calculated that the Fermi level of PCBM is 80 mV higher than that of ICBA. Considering that the LUMO level of PCBM is deeper than ICBA (Fig. 4b), the higher Fermi level of PCBM indicates that more states are occupied by electrons, i.e., it has higher electron density\textsuperscript{27}. This can be attributed to remote doping from iodide, which can act as donor to inject electrons into ETL (ref. 30). In contrast, the shallow LUMO level of ICBA prohibits electron injection from iodide, giving rise to less electron density. The existence of iodide in ETL can be ascribed to ion migration in perovskite film\textsuperscript{30}.

The high electron density of PCBM could aggravate interface carrier recombination with p-type tin perovskite film\textsuperscript{31}. We hence performed time-resolved photoluminescence (TRPL) decay measurement to test the interface carriers recombination. The TRPL of different perovskite films shows a bi-exponential decay (Fig. 4c), including a fast component ($\tau_1$), and a slow component ($\tau_2$)\textsuperscript{32} (Table 2). The fast component can be ascribed to the rapid carriers recombination across the interfaces of perovskite and transporting layer\textsuperscript{33}. The perovskite film contacting PCBM shows a smaller $\tau_1$, indicating faster carriers recombination at the interface. We used electroluminescence (EL) measurement under bias to estimate the interface recombination further\textsuperscript{34,35}. As shown in Fig. 4d, PEA15-SCN device with ICBA shows an obvious EL peak, while extremely weak EL peak is observed for device with PCBM. The much higher EL intensity shows that the interface recombination is significantly suppressed, agreeing well with TRPL characterization.

We then used a model to quantitatively calculate the voltage loss due to interface carrier recombination. The $V_{OC}$ loss ($\Delta V_{OC}^{\text{rad}}$) of non-radiative decay at perovskite/ETL interface can be estimated by external radiative efficiency (ERE) from EL measurement based on the following equation\textsuperscript{36}:

$$\Delta V_{OC}^{\text{rad}} = -\frac{k_B T}{q} \ln \text{ERE}$$

Where ERE is the electroluminescent EQE of the device. Based on the ERE of PEA15 and PEA15-SCN (Supplementary Fig. 16), the $\Delta V_{OC}^{\text{rad}}$ are determined to be 417 mV and 226 mV for the device based on PCBM and ICBA, respectively. Hence, we conclude that interface recombination is another important factor that responsible for the low voltage of the device based on PCBM.

To investigate the function of the addition of NH$_4$SCN in the film to device performances, we performed transient photovoltage (TPV) and transient photocurrent (TPC) measurements to test the density of defect states in the bandgap directly\textsuperscript{37} (Fig. 4e). The film with NH$_4$SCN shows less density of defect states inside the bandgap, especially in the region close to the conduction band. We tested $V_{OC}$ versus illumination intensity curves (Fig. 4f) and found that slope for the device based on PEA15-SCN is smaller than that based on PEA15, indicating the decrease of recombination of the film. Furthermore, the low Urbach energy of PEA15-SCN film in absorption spectra indicates the decrease of defect density as well. The reduced defect density of the film can be

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**Table 1 Summary of the best performance devices with ICBA.**

| Device                  | PCE (%) | $V_{OC}$ (V) | $J_{SC}$ (mA cm$^{-2}$) | FF (%) |
|-------------------------|---------|--------------|-----------------|--------|
| PEDOT/PEA15/ICBA       | 10.1    | 0.78         | 17.8            | 72     |
| PEDOT/PEA15-SCN/PCBM   | 7.7     | 0.60         | 17.1            | 74     |
| PEDOT/PEA15-SCN/ICBA   | 12.4    | 0.94         | 17.4            | 75     |
ascribed to the increase of domain size that reduced the grain boundary of the film, as discussed above. The reduced density of defects of SCN-treated film brings increased voltage and efficiency of the device.

**Discussion**

In this work, we develop a device structure of tin PSCs using ICBA as ETL. The higher LUMO energy level of ICBA improves the maximum attainable voltage of the device and reduces the remote doping caused interface recombination. Furthermore, the use of NH₄SCN as additive and PEDOT as hole transporting layer for perovskite growth reduces the defect density of the film. As a result, we are able to realize an unprecedentedly high voltage of 0.94 V for tin PSCs. Finally, we achieve a certified PCE up to 12.4%, which is almost 30% higher than the highest efficiency reported. With the open-circuit voltage approaching its theoretic limitation, this work indicates a huge potential for efficiency improvement of tin PSCs.
Methods

Device fabrication. SnI₂, formamidinium iodide (FAI), phenethylammonium iodide (PEAI), and NiOₓ were prepared according to literatures22,23. N,N-dime- thylethylenediamine (DMF), DMSO, chlorobenzene (DMSO), isopro- pyl alcohol, SnF₂, and NH₄SCN were purchased from Sigma-Aldrich. Cleaned indium tin oxide (ITO) glass was treated in an ultraviolet ozone (UVO) machine for 20 min before fabrication. The PEDOT (Heraeus-Clevios P VP AI 4083) solution was spin coated onto ITO substrate at 6000 rpm for 60 s and then annealed at 140 °C for 20 min. A 0.8 M perovskite precursor (PEAIFAI5SnI₃SnF₂ = 0.15:0.85:0.0:0.075) and with or without NH₄SCN (5 mol%) were added in mixed solvent (DMF:DMSO = 4:1 V/V) and stirred at 70 °C for 1 h. The precursor was spin coated at 1000 rpm and 5000 rpm for 10 s and 30 s, respectively. And 600 μL toluene (Sinopharm Chemical Reagent Co. Ltd., redistilled) was dropped during the second process at eighth minute. The substrate was then annealed at 70 °C for 10 min. The films of PEAI and PEAI-PEA9 were made by the same procedure. A total of 18 μg mL⁻¹ ICl (1- Material) or PCBM (nano-C) in chlorobenzene was spin coated at 1000 rpm for 30 s, and then annealed at 70 °C for 10 min. Saturated bathocuproine in isopropl alcohol was spin coated at 6000 rpm for 30 s, followed by annealing at 70 °C for 10 min. All precursors were filtered with 0.22 μm polytetrafluoroethylene (PTFE) filters before spin coated. Finally, 100 nm Ag layer was deposited under vacuum of <10⁻⁷ torr using an Angstrom Engineering deposition system.

Film characterization. XRD was recorded on a Bruker D8 Advance using Cu Kα source (λ = 1.54 Å). PL spectra were carried out by exciting the wavelength at 480 nm of spectrophotometer (Fluorolog; HORIBA FL-3) with a standard 40 W Xenon CW lamp. TRPL spectroscopy at 840 nm was measured using Fluorolog spectrophotometer (Fluorolog; HORIBA FL-3) with a pulsed source at 504 nm DeltaDioD DD-5/01. The UV–vis spectra of the perovskite films deposited on PEDOT were recorded by UV–vis spectrophotometer (Agilent Cary5000). SEM images were recorded on JSM-7800. The curves were corrected based on the C1s peak at 284.8 eV. GIWAXS studies were performed at the BL16B1 beamline of Shanghai Synchrotron Radiation Facility (SSRF), Shanghai, China, using beam energy of 10 keV (λ = 1.24 Å) and a Mar 225 detector. The grazing-incidence angles for all materials (including both lattice parameters and internal coordinates) or slab structures, only the internal coordinates are relaxed. The optB86b-vdW (ref. 33) functional is adopted throughout the whole calculation.

Device characterization. J–V curves were measured using a Keithley 2400 source unit under simulated AM 1.5G solar illumination at 100 mW cm⁻² (1 sun). The light intensity was calibrated by means of a KG–5 Si diode with a solar simulator (Enli Tech, Taiwan). The devices are measured in reverse scan (1.0–0 V, step 0.01 V) and forward scan (0–1.0 V, step 0.01 V) with a delay time of 30 ms. The J–V curves for all devices were measured by masking the active area using a metal mask with an area of 0.04 cm². The steady-state PCE was performed at 0.81 V. Devices were stored and tested in the same nitrogen-filled glovebox (Vigor). The EQE spectra were measured by a commercial system (Solar cell scan 100, Beijing Zolix Instruments Co., Ltd). The cells were subjected to monochromatic solar illumination (150 W Xe lamp passing through a monochromator and appropriate filters). The light intensity was calibrated by a standard photodetector (QE/B3/ S1337–1010BQ, Zolix). The light beam was chopped at 180 Hz, and the response of the cell was acquired by a Stanford Research SR830 lock-in amplifier. TPV and TPC measurements were carried out using a setup comprising a 532 nm wavelength laser to provide steady-state bias light, a 640 nm wavelength laser, and an oscilloscope. A power-adjustable 532 nm wavelength laser was used to get a steady VOC of the device. A 640 nm diode laser was used to modulate the VOC on top of a constant light bias. The pulse duration was set to 1 μs and the repetition rate to 50 Hz by the function generator of the oscillator. The digital oscilloscope recorded the data induced by the light perturbation, using a 1 MHz input impedance for the TPV measurement and 50 Ω impedance for TPC measurement. We determined the charge generated (AQ) in the devices by integrating the TPC curve by the 640 nm laser pulse without 532 nm light bias present. The calculated C ≡ AQ/VOC is the capacitance. The total charge carrier was obtained by integrating the C versus VOC. The carrier concentration (n) for each open-circuit voltage was calculated by dividing each charge carriers with the device volume. The density of states (DOS) in the mid-gap can be obtained by differentiating the carrier density with respect to the VOC following a previously reported procedure27. The current density–luminance–radiance (J–V–L) characteristics were measured by a Keithley 2400 source meter, and a fiber integrating sphere (FOIS)-couple with a QE Pro 650 spectrometer (Ocean Optics). The devices were tested on top of the integrating sphere, and only forward light emission could be collected. All device test processes were carried out in the N₂-filled glovebox.

First-principles calculations. First-principles calculations were performed using the framework of DFT using plane-wave pseudopotential methods, as implemented in the Vienna Ab-initio Simulation Package39. The generalized gradient approximation formulated by Perdew, Burke, and Ernzerhof was used as the exchange–correlation functional. The electronic–core interactions were described by the projector augmented-wave40 method for the pseudo potentials. The cutoff energy for the plane-wave basis set used to expand the Kohn–Sham orbitals was 400 eV. The Gamma-centered k-point mesh with a grid spacing of 2π × 0.03 Å⁻¹ was used for electronic Brillouin zone integration. For adsorption energy calculation, the vacuum thickness was set to be 20 Å. The equilibrium structural parameters (including both lattice parameters and internal coordinates) of each involved bulk material were obtained via total energy minimization by using the conjugate gradient algorithm, with the force convergence threshold of 0.01 eV Å⁻¹, for slab structures, only the internal coordinates are relaxed. The optB86b-vdW (ref. 33) functional is adopted throughout the whole calculation.

SKPM measurement. The amplitude-modulation SKPM was operated combined with a Cypher S atomic force microscopy (AFM; Asylum Research, Oxford Instruments) and a HF2LI Lock-in amplifier (Zurich Instruments) in N₂-filled glovebox. The resonance frequency ω₀ and spring constant of AFM conducting tips are ~127 kHz and 5.0 Nm⁻¹, respectively.

Tin PSC certification at SIMIT (Shanghai, China). The device certification tests were performed at an independent lab (SIMIT, Chinese Academy of Sciences). And SIMIT is accredited by China National Accreditation Service for Conformity Assessment (CNAS) to ISO/IEC 17025 and by the International Laboratory Accreditation Cooperation (ILAC) Mutual Recognition Arrangement. A silicon reference solar cell (PVM1211, NREL_ISO tracking-1974) was used to set the irradiance at 100 mW cm⁻² at standard testing conditions in accordance with IEC 60904-3 ed.2 AM 1.5 G. J–V characteristics of tin PSCs were measured under simulated sunlight by steady-state class AAA solar simulator according to IEC 60904-9 ed.2. The spectral mismatch was calculated and mismatch correction was performed according to IEC 60904-7 ed.3. The J–V curves were measured in forward and reverse scans with a scanning speed of 90 mV s⁻¹.

Encapsulation method. Devices were encapsulated by quartz glass and UV-glu for J–V, EQE, TRPL, TPV, and TPC measurements. Films were double sealed by plastic bags before XRD, SEM, PL, UV–vis, and GIXWAXS measurements.

Reporting summary. Further information on experimental design is available in the Nature Research Reporting Summary linked to this paper.

Data availability

The data that support the findings of this study are available from the corresponding author upon reasonable request.

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Author contributions

X.J. and F.W. contributed equally to this work. X.J., F.W., and Z.N. designed and directed the study. X.J. and F.W. contributed to all the experimental work. Q.W. carried out and interpreted the DFT studies. H.L. performed the PL, TRPL, and XRD measurements. Y.S. helped to carry out GIWAXS, and EL measurements and data analysis. W.Z. supervised the EQE and TPV measurements. P.C. supervised the UPS measurement. C.W., Q.C., and L.C. conducted the SKPM measurement and data analysis. Z.N. supervised the whole project. X.J., F.W., and Z.N write the manuscript. All authors discussed the results and commented on the final manuscript.

Competing interests

The authors declare no competing interests.

Additional information

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Correspondence and requests for materials should be addressed to Z.N.

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