Deeply-sourced formate fuels sulfate reducers but not methanogens at Lost City hydrothermal field

Susan Q. Lang1,2, Gretchen L. Früh-Green2, Stefano M. Bernasconi2, William J. Brazelton3, Matthew O. Schrenk4 & Julia M. McGonigle3

Hydrogen produced during water-rock serpentinization reactions can drive the synthesis of organic compounds both biotically and abiotically. We investigated abiotic carbon production and microbial metabolic pathways at the high energy but low diversity serpentine-hosted Lost City hydrothermal field. Compound-specific 14C data demonstrates that formate is mantle-derived and abiotic in some locations and has an additional, seawater-derived component in others. Lipids produced by the dominant member of the archaeal community, the Lost City Methanosarcinales, largely lack 14C, but metagenomic evidence suggests they cannot use formate for methanogenesis. Instead, sulfate-reducing bacteria may be the primary consumers of formate in Lost City chimneys. Paradoxically, the archaeal phylotype that numerically dominates the chimney microbial communities appears ill suited to live in pure hydrothermal fluids without the co-occurrence of organisms that can liberate CO₂. Considering the lack of dissolved inorganic carbon in such systems, the ability to utilize formate may be a key trait for survival in pristine serpentine-hosted environments.

Deep sea hydrothermal systems have long been recognized as important zones of biological activity in the anoxic subsurface, and as potential locations for the evolution of early life on Earth1,2. Alkaline serpentinization environments have been proposed as promising locations for the emergence of early life, in part because the high concentrations of H₂ that are a natural outcome of water-rock reactions can promote the non-biological synthesis of some organic molecules3. These environments are widespread in oceanic and continental settings4, may have been even more extensive prior to ~3 billion years ago5, and may be active on other planetary bodies such as Saturn’s moon Enceledus6. While the chemical energy for the synthesis of organic compounds is abundant in serpentinization systems, the relative importance of abiotic vs. biotic processes is unknown. We carried out studies at an alkaline serpentinization system to investigate the abiotic production of organic molecules and to identify how present-day microbial communities interact with these geologically-driven reactions.

The reaction of ultramafic rocks with water results in high concentrations of hydrogen that can provide the geochemical fuel for both biotic and abiotic organic synthesis. Resulting fluids contain high concentrations of reduced carbon compounds such as methane, short chain n-alkanes, and formate that have been proposed to form abiotically based on thermodynamic calculations, laboratory experiments, and geochemical and isotopic measurements7–15. Pure hydrothermal fluids can support anaerobic metabolisms such as methanogenesis and sulfate reduction in the absence of mixing with seawater16–18.

The oxidized carbon species CO₂ or its other forms (HCO₃⁻, CO₃²⁻) are typically presumed to be the starting material for the abiotic formation of organic molecules and for primary production. However, oceanic low-temperature alkaline serpentinization environments such as the Lost City Hydrothermal Field, or analogous land-based environments in Oman, California, and Italy, are characterized by extremely low concentrations of dissolved inorganic carbon due to its reduction to hydrocarbons and the rapid precipitation of calcium carbonate at pH above ~917,19–23. Thus, the availability of oxidized carbon may be a limiting factor for both abiotic
organic carbon synthesis and for the habitability of serpentinization environments. The organic acid formate has been proposed as an alternative starting material. Benchtop experiments have demonstrated that formate can rapidly form under high H₂, reducing conditions, and that certain abiotic synthesis reactions proceed more readily when formate is used as a starting material.

The Lost City Hydrothermal Field is an iconic example of a low-temperature serpentinization system, and fluids there have elevated formate concentrations (36–158 µM) that have been proposed to form abiotically, through equilibration:

\[ \text{HCO}_3^- + \text{H}_2 = \text{HCOO}^- + \text{H}_2\text{O} \]  

(1)

Formate has also been proposed to form abiotically in the lower pH system of Mid-Cayman Rise, but only after mixing of the hydrothermal fluids with local seawater. If mixing is also required for formate production at Lost City, neither it nor \( \sum \text{CO}_2 \) (sum of CO₂aq, H₂CO₃, HCO₃⁻, and CO₃²⁻, used to distinguish from the chemically identical but isotopically distinct seawater dissolved inorganic carbon) would be available as a starting material for abiotic synthesis or primary production during much of the hydrothermal fluid pathway, potentially limiting the importance of alkaline serpentinization in the development of early life.

We carried out a series of ¹³C and ¹⁴C analyses to identify the origin of formate at Lost City and to determine if this compound represents an important carbon source to the dense microbial communities that inhabit the Lost City chimneys. The anoxic interior of chimneys are dominated by a single archaeal phylotype, the Lost City Methanosarcinales (LCMS). Endmember fluids retain 1–4 mM sulfate that co-exists with hydrogen, and geochemical trends across the field indicate microbial sulfate reduction is an important subsurface process, possibly carried out by organisms related to Desulfotomaculum. Chimney exteriors host organisms involved in the oxidation of sulfur and CH₄ (e.g. Methylomonas, Thiomicroscopica).

Results and Discussion

The isotopic signature (¹³C, ¹⁴C) of formate was determined for 6 fluid samples collected from Lost City chimneys at 4 locations (Markers 2, C, B, 3; Fig. 1, Table 1 and Supplemental Table 1). Formate from Marker 2 lacks radiocarbon (F¹⁴C = 0.09 ± 0.09), demonstrating that it is ultimately derived from ¹⁴C-free carbon (Fig. 2; Supplemental Table 1). It eliminates the fermentation or degradation of chimney biomass as a possible source, as chimney total organic carbon has F¹⁴C values ranging from 0.480 to 0.888 (Table 1). The stable isotope signatures of formate from both Marker 2 and Marker C (−13.0 to −8.9%) overlap with those of ¹⁴C-free carbon (Fig. 2; Supplemental Table 1). Mantle CO₂ is inferred to be the starting material for C₁⁺ alkanes; although this CO₂ is subsequently removed from the fluids due to abiotic reduction to organic compounds and/or precipitation as carbonate. Input of mantle volatiles is also evident from elevated ³He/⁴He ratios. The lack of ¹⁴C in the formate indicates it is also formed from mantle CO₂. Temperatures in the subsurface of Lost City (~180 °C) are lower than the current temperature limit to life (122 °C). In contrast, formate from Markers B and 3 was present in higher concentrations, was somewhat modern (F¹⁴C = 0.14–0.56),
and had more positive δ^{13}C values (−9.1 to −4.3%) (Fig. 2; Supplementary Table S1). The more modern formate at Markers B and 3 points to an additional carbon source in these locations.

The relative concentrations of formate and ΣCO₂ in Lost City fluids suggest they are not controlled by purely equilibrium processes. In almost all locations at Lost City, formate is present in far higher concentrations than expected for the amount of ΣCO₂ present (ref.14; Supplemental Materials). The one exception is at Marker 3 where the two species are close to equilibrium due to ΣCO₂ concentrations that are more than an order of magnitude higher than those at other vents (10–26 versus 0.1–0.6 µmol/L)13. Fluids from Marker 3 have been influenced by near-surface mixing with seawater that is the likely source of the additional inorganic carbon14,34.

In the simplest scenario, formate has two sources: an abiotically synthesized, mantle-derived, F^{14}C free source with a more negative δ^{13}C signature and an additional source derived from modern inorganic carbon in seawater.

### Table 1. Concentrations and isotopic compositions of known carbon species in the Lost City environment.

| Species                                | Concentration (µmol/L) | δ^{13}C (% vs VPDB) | F^{14}C |
|----------------------------------------|------------------------|---------------------|--------|
| **Constituents dissolved in fluids**   |                        |                     |        |
| CH₄,12,21                              | 890–1980               | −13.6 to −9.3       | 0.002 to 0.006 |
| C₂H₂ - C₄H₁₀,13                        | 0–1.8                  | −16.9 to −13.1      | n.d.   |
| Formate, present work                  | 36–158                 | −12.1 to −6.8       | 0.09 to 0.56 |
| Acetate, present work                  | 1–35                   | −26.6 to −20.3      | n.d.   |
| Total hydrolizable amino acids         | 0.7–2.3                | n.d.                | n.d.   |
| Mantle CO₂, predicted to be in endmember fluids* | 2000–4100              | −12 to −2          | 0      |
| Dissolved ΣCO₂ in endmember fluids from Mkrs C, 2, B13 | 0.1–1 (avg 0.2)        | ≈ −9               | n.d.   |
| Dissolved ΣCO₂ in endmember fluids from Mkr 313 | 10–26                  | ≈ −9               | n.d.   |
| Dissolved inorganic carbon in seawater (800 & 550 m)13 | 2200                   | 0.8                | 0.993 to 1.052 |
| **Constituents in actively venting chimneys** |                        |                     |        |
| Calcium Carbonate,21,25                 | n/a                    | −7 to +13           | 0.957 to 1.002 |
| Total organic carbon12,25               | 400–1500 ppm           | −12.4 to −3         | 0.480 to 0.888 |
| Total hydrolizable amino acids27        | n.d.                   | −18.3 to +8.7       | n.d.   |
| Bacterial lipids (fatty acids, non-isoprenoidal diethers)23,26 | 0–2.4 µg lipid/g chimney | −31 to −1.1 | n.d. |
| Archaeal lipids (isoprenoidal diethers, PMIs)37,42 | 0–3.5 µg lipid/g chimney | −12 to +24.6 | n.d. |
| Eukaryotic lipids (polycyclic terpanoids)37,13 | 0–0.5 µg lipid/g chimney | −28.5 to −15.4 | n.d. |
| Mono-unsaturated fatty acids, >95% C₁₆:1 and C₁₈:1b | −18.7                 | 0.68 ± 0.03         |        |
| Saturated, straight-chain fatty acids, >99% C₁₆:0b | −8.9                  | 0.13 ± 0.02         |        |
| Combined Pentamethylicosanesb,c | 1.7                    | 0.24 to 0.49        |        |
| Squaleneb                              | −11.0                  | 0.17 ± 0.02         |        |
| Phytanic Acidb                         | −1.4                   | 0.22 ± 0.02         |        |

Figure 2. Average (a) F^{14}C of formate vs. formate concentration and (b) δ^{13}C of formate vs. formate concentration in different vents. Error bars represent the standard deviation of multiple samples (N = 2–6; full data in Supplemental Table 1). The amount of formate isolated from Marker C samples was insufficient for F^{14}C analysis.

The relative concentrations of formate and ΣCO₂ in Lost City fluids suggest they are not controlled by purely equilibrium processes. In almost all locations at Lost City, formate is present in far higher concentrations than expected for the amount of ΣCO₂ present (ref.14; Supplemental Materials). The one exception is at Marker 3 where the two species are close to equilibrium due to ΣCO₂ concentrations that are more than an order of magnitude higher than those at other vents (10–26 versus 0.1–0.6 µmol/L)13. Fluids from Marker 3 have been influenced by near-surface mixing with seawater that is the likely source of the additional inorganic carbon14,34.

In the simplest scenario, formate has two sources: an abiotically synthesized, mantle-derived, F^{14}C free source with a more negative δ^{13}C signature and an additional source derived from modern inorganic carbon in seawater.
metagenomes were most similar to a member of family Peptococcaceae (order Clostridiales) that can use formate as an electron donor \(^3\). However, City chimneys and fluids \(^2\,\,\,^4\,\,\,^3\,\,\,\,\,^4\). Contributions of a modern endmember with an isotopic signature of F\(^{14}\)C = 1.02 ± 0.04 and F\(^{13}\)C = + 8.4 ± 1.6% and a concentration that varies between locations, from 3–40 \(\mu\)mol/L. The derivation of these values is described in full in the Supplemental Materials.

The calculated concentration of formate that is radiocarbon-free at Marker 3 is significantly less (72 ± 8 \(\mu\)mol/L), while the calculated concentration of modern-derived formate is significantly more (−90 \(\mu\)mol/L) than in the other chimneys. This apparently lower contribution of \(^{14}\)C-free formate at Marker 3 is remarkable given that its total concentration is higher than the other locations. When combined with the observation that the concentrations of formate and \(\Sigma\)CO\(_2\) are close to equilibrium in this location, the F\(^{14}\)C data point to a rapid equilibration of both concentrations and isotopes between these two species at Marker 3. This equilibration is, presumably, due to biological processes as the abiogenic equilibration between the two is expected to be exceedingly slow at temperatures <175 \(^\circ\)C and fluid temperatures are 88 \(^\circ\)C. \(^5\,\,\,^9\,\,\,^3\,\,\,^3\,\,\,\,\,^3\). In contrast, the biological conversion between CO\(_2\) and formate is rapid and reversible. \(^9\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\). Therefore, microbial communities living along the fluid pathway of Marker 3 may be rapidly converting between the mantle-derived, \(^{14}\)C-free formate carried with pure endmember fluids and the inorganic carbon that is present in the seawater that has been entrained. If true, the \(^{14}\)C of the \(\Sigma\)CO\(_2\) from Marker 3 should be a similar mixture of modern and dead carbon. Unfortunately, it has not been possible to make this measurement due to low radiocarbon concentrations \(^1\). This same process likely occurs at other locations but is less evident due to a lower amount of seawater entrainment.

To investigate whether the microbial communities inhabiting the carbonate chimneys are actively utilizing formate, we determined the \(^{14}\)C content of lipid biomarkers from a Lost City chimney (Marker 7) that could be attributed to specific microbial communities. Previous work has demonstrated that bacterial fatty acids have a bimodal \(^{13}\)C distribution in many chimneys, with fully saturated C16:0 and C18:0 having more positive isotopic signatures than unsaturated molecules (e.g. −25.4 to −16.8 versus −9.1 to −1.1 respectively at Marker 7) \(^4\). This pattern has been attributed to a distinction between bacteria inhabiting the interior of the chimney where seawater DIC availability is limited and those living on the exterior of the chimney where it is abundant \(^8\). In agreement with this interpretation, the saturated C16:0 fatty acid lacks \(^{14}\)C while, in contrast, the mono-unsaturated fatty acids C16:1 and C18:1 have a mixed F\(^{14}\)C signature of 0.68 ± 0.03 (Table 1). This result demonstrates that bacteria living in the anoxic interior of the chimney metabolize a \(^{14}\)C-free carbon source that is consistent with their utilization of formate. Mantle CO\(_2\) is also \(^{14}\)C-free but multiple lines of evidence indicate it is stripped from fluids in the subsurface: (a) similar to other alkaline serpentinitization systems \(^5\), Lost City endmember fluids lack significant concentrations of any \(\Sigma\)CO\(_2\) (0.1–0.6 \(\mu\)mol/L, with the exception of Marker 3 where seawater inputs raise concentrations to 10–26 \(\mu\)mol/L, Table 1) \(^4\) due to the rapid precipitation of calcium carbonate in the subsurface \(^2\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\). (b) Calcium carbonate from actively venting chimneys has modern \(^{14}\)C signatures, indicating that it is precipitated from seawater bicarbonate, and that mantle CO\(_2\) is not available (Table 1) \(^2\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\,\,\,\,\,^3\). Metagenomic data from two Lost City chimneys (Marker 3, Marker 5) are also consistent with the ability of some chimney microorganisms to metabolize formate. The enzyme formate dehydrogenase catalyzes the reversible oxidation of formate to CO\(_2\). \(^4\). Several sequences of the alpha subunit of formate dehydrogenase (fdhA) were detected in the chimney metagenomes; the closest bacterial database matches were with sequences related to Desulfotomaculum and that these organisms may be fueled by formate instead of or in addition to hydrogen.

The archaeological community in the chimneys is dominated by the Lost City Methanosarcinales (LCMS), to the point that they are the only detectable archaea in some locations \(^2\,\,\,^4\). Positively \(^{14}\)C enriched lipid biomarkers that are exclusively produced by methane-metabolizing archaea, pentamethylicosane and its unsaturated related compounds (PMIs), have been attributed to carbon limitation by methanogens consuming dissolved inorganic carbon \(^5\,\,\,^4\). Attempts to fully isolate the PMIs for \(^{14}\)C analysis were unsuccessful as they co-elute with a minor unresolved complex mixture (UCM) hump. The total F\(^{14}\)C value of the combined isolate was F\(^{14}\)C = 0.39 ± 0.02 (Table 1). Mass balance calculations assuming the UCM is either entirely modern or entirely dead require the PMIs to have a F\(^{14}\)C value of 0.24–0.49. Phytanic acid and squalene have also been detected in the carbonate chimneys and tentatively attributed to the same archaeological community based on their \(^{13}\)C signatures, though contributions from bacterial or eukaryotic sources cannot be ruled out \(^4\). These compounds have F\(^{14}\)C values that are 0.22 ± 0.02 and 0.17 ± 0.02, respectively (Supplemental Table 2). The mixed \(^{14}\)C signature of PMIs rules out that the LCMS depends entirely on seawater dissolved inorganic carbon or materials, such as the calcium carbonate of the chimneys, derived from seawater DIC. Instead, it must rely on a carbon source with a substantial mantle-derived component but, given the mixed \(^{14}\)C signature, it is also unlikely that they solely...
Neither of the two archaeal phylogeny of genes from methanogens encoding sequences was constructed (Supplementary Figure 2). fdhA was highly divergent from all known fdhA from methanogens known to use formate as a carbon source. The predicted archaeal formate dehydrogenase gene. This interpretation is consistent with a recent study that identified Bay in New Caledonia) while methane cycling archaea are only identified in older, more established chimneys.

Archaeal formate dehydrogenase sequences of the fdhA enzyme were also identified in the metagenomic data (Table 2). To investigate whether these represent methanogenic archaea utilizing formate as a carbon source, a phylogeny of genes from methanogens encoding fdhA sequences was constructed (Supplementary Figure 2). Neither of the two archaeal fdhA sequences in Table 2 showed close phylogenetic relationships with fdhA genes from methanogens known to use formate as a carbon source. The predicted archaeal fdhA sequence from Marker 3 was highly divergent from all known fdhA sequences and was not included in Supplementary Figure 2.

The second archaeal fdhA sequence from Marker 5 (large bold font in Supplementary Figure 2) belongs to a divergent clade of fdhA that includes two species of Methanolobus that can only use methylated compounds as methanogenic substrates and a metagenomic sequence from an enrichment of ANME-2 anaerobic methanotrophic archaea. (ANME-2 and Methanolobus species belong to the same order as the Lost City Methanosarcinales.) These four fdhA sequences form a clade that is clearly distinct from any fdhA sequences from known formate-utilizing or hydrogenotrophic methanogens (Supplementary Figure 2).

The presence of fdhA is necessary, but not sufficient, to enable formate-based methanogenesis. Many methanogens that contain fdhA are not able to use formate as a carbon source (Supplemental Figure 2). The inability of many of these methanogens to utilize formate is mysterious, but the gene fdhC, which is predicted to encode a transporter protein, is thought to be necessary for transport of formate into the cell. A search for fdhC in the Lost City chimney metagenomes, recovered only proteins similar to Thiomicrospira transporters that are more likely to be transporters of nitrite, not formate. No fdhC sequences belonging to methanogenic archaea were detected. The absence of genes in metagenomic data must always be interpreted with caution, but the lack of fdhC sequences in this dataset supports the interpretation of the fdhA phylogeny that Lost City Methanosarcinales archaea are unlikely to directly utilize formate from the environment.

Ideally, all three lines of evidence presented here would have been carried out on samples from the same suite of Marker locations. The vast majority of samples from Lost City have been previously consumed during earlier analyses. Here we have applied techniques that have been developed in the >12 years since these samples were collected to the few that remained in our laboratories. In many cases, however, extrapolating between locations is justified by results from previous studies. For example, while the lipid 14C data is solely from Marker 7, previous work indicates this location has fatty acid and archaeal lipids with similar concentrations, distributions, and 13C signatures similar to Markers 2, 3, B, H, and C. Similarly, while the metagenomic data is from Markers 3 and 5, 16S rRNA data indicates the dominance of the same Lost City Methanosarcinales phylotype in all analyzed actively venting chimneys (Markers 2, 3, H, C).

The picture that emerges is a system in which abiotic formate is consistently supplied with pure hydrothermal fluids and utilized by sulfate reducing bacteria who, in part, convert it to \( \text{CO}_2 \) (Fig. 3). This CO2 would briefly be available to other members of the anaerobic microbial community, including the Methanosarcinales. In near-surface regions the \( \text{CO}_2 \) pool would mix with minor inputs of modern seawater DIC, either from entrainment from seawater or the minor dissolution of the chimneys themselves. The mixed 14C signature of the PMIs could therefore be accounted for by the Methanosarcinales consuming CO2 liberated during the activity of the formate dehydrogenase gene. This interpretation is consistent with a recent study that identified Firmicutes as the first inhabitants of juvenile and nascent chimneys in an analogous alkaline serpentinization environment (Prony Bay in New Caledonia) while methane cycling archaea are only identified in older, more established chimneys.

Serpentinization systems have been increasingly proposed as locations where early metabolic pathways could have emerged on Earth and other places, and they are also critical to understanding the extent and activity of life below the oxygenated surface of Earth. Abundant geochemical energy is available to drive abiotic synthesis and/or microbial metabolism, but the lack of dissolved inorganic carbon may limit the reactions that can proceed. The last universal common ancestor (LUCA) has been proposed to be an anaerobic autotroph similar to methanogens and clostridia, living off of \( \text{H}_2 \), \( \text{CO}_2 \), and \( \text{N}_2 \) in an environment such as Lost City.

### Table 2. Metagenomic sequences predicted to encode the alpha subunit of formate dehydrogenase (fdhA) in Lost City chimneys, as defined by KEGG protein function K00123.

| Location | Contig ID | Start - Stop | Best Match | Identities | Metagenome Coverage |
|----------|-----------|--------------|------------|------------|---------------------|
| Marker 3 | Marker 3 3862 | 3862contig:30000000 | 319–1782 | ANME-2 metagenome (CAI64341.1) | 50% 80.74 |
| Marker 3 | Marker 3 3862 | 3862contig:19000000 | 620–919 | Desulfobacter alkalitolerans (WP_028308877.1) | 50% 78.59 |
| Marker 3 | Marker 3 3862 | 3862contig:70000000 | 76–252 | Methylophaga muraisi (WP_009296139.1) | 89% 25.59 |
| Marker 5 | Marker 5 H08contig | 30000000 | 224–1687 | ANME-2 metagenome (CAI64341.1) | 50% 107.85 |
| Marker 5 | Marker 5 H08contig:0 | 245–823 | Desulfobacter alkalitolerans (WP_028308876.1) | 74% 72.38 |
| Marker 5 | Marker 5 H08contig:0 | 959–1291 | Desulfotalea psychrophila (WP_041277585.1) | 55% 72.38 |
| Marker 5 | Marker 5 H08contig:8000000 | 101–232 | Thalassobius mediterraneus | 106% 12.21 |
Materials and Methods

Sample Collection. Fluids from eight actively venting carbonate-brucite chimneys across the Lost City hydrothermal field were collected in 2003 by DSV Alvin using previously described ‘major’ samplers \(^{46}\) and the Hydrothermal Fluid and Particulate Sampler (HFPS) \(^{47}\). These samplers are well-suited for the analysis described here because they have low organic carbon blanks when properly cleaned \(^{48}\). Both filtered (in situ) and unfiltered HFPS samples were collected; filters used in situ were either glass fiber filters (Whatman GF/F; combusted at 500 °C for 5 h) or 0.2 µm nitrocellulose filters (Millipore). Upon arrival on deck, subsamples for organic analyses were stored in 100 mL glass vials (combusted at 500 °C, 5 h) with acid-washed Teflon lined caps (10% HCl, overnight); samples were then stored frozen (20 °C) until analysis on shore.

The chimney samples used for isotopic and metagenomic analyses were collected during two cruises. Sample 3862–1325 was collected from Marker 3 in 2003 by DSV Alvin aboard the R/V Atlantis. Samples H08_080105_Bio5slurpB1, from Marker 5, and H02_072605_Bio1slurpA2_0354, from Marker 7, were collected in 2005 during a National Oceanic and Atmospheric Administration (NOAA) Ocean Explorer cruise with the ROV Hercules aboard the R/V Ronald H. Brown. In all cases, samples were immediately placed in a sterile Whirl-Pak® sample bag upon arrival on deck and stored at −80 °C until analysis. DNA was extracted from the Marker 3 and Marker 5 samples according to a previously published protocol \(^{49}\). Lipids were extracted from the Marker 7 sample as described below. Geochemical analysis of sample 3862-1325 was previously published \(^{50,51}\). No previous microbiological analyses of this sample have been published, but a similar sample collected from Marker 3 during the same expedition (3881-1408) was the subject of previously published DNA sequencing studies \(^{28,40}\), where it was referred to as LC-7a and LC-5, respectively.

Compound Specific Isotope Analysis of Organic Acids. The isotopic signatures of formate and acetate were determined with the method of \(^{52,53}\). Briefly, the method consisted of the following steps: (1) Apolar material is removed from 5–15 mL samples by passing them over a C18 SPE cartridge; (2) Sample pH is adjusted to > 9 with NaOH and concentrated by freeze-drying; (3) Organic acids are separated by HPLC; (4) Fractions are collected in sealed 12 mL Exetainer® screw capped vials that had been previously spiked with a chemical oxidant and purged with helium; (5) oxidation of the organic acids to CO\(_2\) is achieved by heating the samples to 90 °C for 15 minutes; (6) the δ\(^{13}\)C value of the CO\(_2\) is determined by injection into a gas chromatograph coupled to an isotope ratio mass spectrometer (Agilent 6890 gas chromatograph equipped with a CP Poraplot Q column (27.5 m x 0.32 mm, 10 µm, Varian) maintained at 100 °C, interfaced with a ConFlo IV to a Delta V Plus Mass Spectrometer, both ThermoFisher Scientific); and (7) the Δ\(^{14}\)C value of the remaining CO\(_2\) is determined by Accelerator Mass Spectrometry (AMS) at the Institute of Particle Physics of the ETH Zürich using the gas ion source of the 200 kV using a Mini Carbon Dating System (MICADAS) \(^{54}\).

Compound Specific Isotope Analysis of Lipid Biomarkers. Sample H02_072605_Bio1slurpA2_0354 was sequentially extracted three times each with methanol, a 1:1 methanol:dichloromethane mixture, and dichloromethane (DCM). The extracts were combined and the solvent removed under vacuum. The total lipid extract
was saponified with 0.5 M KOH in methanol for 2 hours at 70 °C. After cooling, the mixture was back extracted with hexane three times to isolate the neutral fraction, then acidified to pH 2 with 12 M HCl and back extracted three times with hexane:DCM (8:2) to isolate the acid fraction. Identification, quantification, and final checks on compound purity of isolated fractions were performed on a Hewlett Packard 6890 Series gas chromatography system equipped with a mass selective detector (Hewlett Packard 5973).

The acid fraction was methylated overnight at 45 °C with 1.2% HCl in methanol. Phthalic acid of known isotopic composition was simultaneously methylated so that the addition of the –CH₃ groups could be accounted for in reporting final isotope values. The fatty acid methyl esters (FAMES) were separated over silver impregnated silica gel (AgSiO₂) with 100% hexane, 100% DCM, and Hexane:ethyl acetate (98:2). Unsataturated fatty acids eluted in the hexanexethyl acetate fraction and contained primarily (>95%) 16:1 and 18:1, with smaller contributions of other unsaturated fatty acids. This fraction was prepped for radiocarbon analysis without further purification. The DCM fraction contained saturated fatty acids, which were further separated with urea adduction into branched (non-adducted) and unbranched (adducted) compounds. Both fractions were further purified over silica gel to eliminate any traces of urea. The non-adducted fraction contained solely phytanic acid while the adducted fraction contained straight-chain fatty acids, dominated by C16:0. Both fractions were prepped for radiocarbon.

The neutral fraction was passed over columns of activated copper and sodium sulfate to remove sulfur and water, respectively. The neutral fraction was then separated over 5% deactivated silica gel (pore size 60 Å, 40–63 μm particle size, activated at 120 °C for 4 hr). Polar neutral compounds were eluted with hexane:DCMeOH (1:1) and n-alkanes were eluted with hexane:DCM (9:1). The n-alkane fraction contained saturated and unsaturated pentamethylicosanes (PMIs), squalane, squalene, and variably unsaturated squalenoids. A preparative gas chromatography was used to further separate these compounds. It consisted of an HP 6890 gas chromatograph with a cooled injector syringe (Gerstel GmbH, Mülheim an der Ruhr, Germany) with a megabore fused silica column (DB-XLB, 30 m × 0.53 mm i.d.), a flame ionization detector, and a preparative fraction collector (PFC, Gerstel). Compounds were collected in glass u-traps attached to the PFC. Collection windows were timed to isolate all of the PMIs, the squalenoids, and squalene. The squalenoid fraction was lost; the squalene fraction was pure and was prepped for radiocarbon. The PMIs contained not only the compounds but also a significant UCM hump. To further purify this fraction the PMIs were first hydrogenated to the fully saturated PMI. The fraction was then passed over Ag-SiO₂ gel and the hexane fraction was collected. After these steps, the PMI peak contributed 80% of the total peak area.

Two procedural standards, a 14C-free C26 alkane (Fluka P/N 52185, lot 405359/1, F₁⁴C = 0.0016 ± 0.0006) and a modern C30 alkane (Fluka P/N 90270-1 G, lot 044661/1, F₁⁴C = 0.9918 ± 0.0038) were subjected to similar isolation steps to characterize the contribution of extraneous carbon. For the compounds isolated by AgSiO₂ chromatography and urea adduction (saturated fatty acids, phytanic acid, unsaturated fatty acids), different concentrations of both process standards were subjected to AgSiO₂ chromatography, urea adduction, and a final purification over 5% deactivated silica gel. Because the alkane standards will not elute in the same fraction as more polar the compounds of interest, all eluted fractions were combined as one to obtain a maximum representative blank. A separate series of both standards was also injected into the preparative gas chromatograph, trapped, and processed in the same way as the samples. In both cases, the process standards incorporate any extraneous carbon contributed from the vacuum oxidation and introduction into the AMS.

Isolates to be prepped for radiocarbon were dissolved in DCMeOH (9:1) and passed over a very short silica gel column directly into pre-combusted (950 °C, 5 hours) quartz tubes. After addition of pre-combusted (950 °C, 5 hours) CuO, the tubes were flame sealed under vacuum and the compounds oxidized to CO₂ overnight at 950 °C. The CO₂ was quantified on a vacuum line and transferred for 4 × 70 mm Pyrex tubes for subsequent AMS measurement.

The radiocarbon content was analyzed with a miniaturized radiocarbon dating system (MICADAS) at the laboratory of Ion Physics of the ETH. Results are reported as F₁⁴C. Different amounts of radiocarbon-dead phthalic acid and modern oxalic acid (OX-2) were also placed in quartz tubes and prepared in a similar way as the samples and process standards.

The final F₁⁴C values for the saturated fatty acids, the phytanic acid, and the unsaturated fatty acids were corrected using the extraneous carbon determined by column chromatography and for the addition of the –CH₃ group. The F₁⁴C of squalene was corrected for the extraneous carbon determined by prep-GC. In all cases, the extraneous carbon contributed ≤3% of the total C in the sample. Propagated errors from the blank correction and the AMS analysis are reported.

**Metagenomic sequencing.** Purified DNA extracted from the two chimney samples was sent to the Josephine Bay Paul Center, Marine Biological Laboratory (MBL) for Illumina shotgun sequencing. Metagenomic libraries were constructed with the NuGen Ultralow Ovation kit according to the manufacturer’s instructions. Paired-end sequencing was conducted with a 100 cycle Illumina HiSeq run. All unassembled sequence data related to this study are available via the SRA identifier SRP049438 and BioProject PRJNA265986. Raw sequence data was processed by the Brazelton lab to trim adapter sequences, remove artificial replicates, and trim reads based on quality so that all reads have a minimum quality score of 25 over a sliding window of 6 bases. Paired-end reads from each chimney sample were assembled separately with Ray Meta with a kmer of 41. Several assembly strategies and parameters were tested and compared with MetaQuast and ALE. The Prokka pipeline was used for gene prediction and functional annotation. The annotations–metagenome and–proteins were used with Prokka v.1.12 to indicate that genes should be predicted with the implementation of Prodigal v.2.6.2, optimized for metagenomes, and then searched preferentially against a custom protein database as well as the default databases included with Prokka, including Pfam. The database provided was the
last free version (2011) of the Kyoto Encyclopedia of Genes and Genomes, obtained from MG-RAST. Contig coverages were calculated by mapping merged reads onto the assembled contigs with Bowtie2 v.2.2.6 and then multiplying the number of mapped reads to each contig by the average read length of 108 base pairs and dividing by the length of the contig. Detailed documentation of all metagenomic data processing is provided on the Brazelton lab’s website (https://baas-becking.biology.utah.edu/data/category/18-protocols), and all custom software and scripts are available at https://github.com/Brazelton-Lab.

Analyses of formate dehydrogenase genes. Assembled metagenomic contigs with predicted fdhA sequences, i.e. those that encode the alpha subunit of formate dehydrogenase (K00123), were identified in each chimney metagenome. The best matches to the predicted fdhA genes were identified by conducting a blastx search with each predicted fdhA nucleotide sequence against the NCBI non-redundant protein sequences (nr) database. The best match reported in Table 2 of the main text was that which had the highest ‘max score’ in the blastx search. ‘Max score’ was chosen to more heavily weight matches that included the full length of the protein sequence, rather than matches that only had high similarity over a short region of the protein. Reference protein sequences for the phylogeny of archaean fdhA were obtained from all finished archaean genomes in JGI’s IMG database in July 2016. Predicted fdhA protein sequences from the Lost City contigs were included in the tree, and the final tree in Supplemental Figure 2 was refined to include only species with close phylogenetic relationships with Lost City sequences. The multiple sequence alignment was conducted with MUSCLE, and the phylogeny was built using RaxML and the ‘-f a’ option with 100 bootstrap replicates.

Data Availability. Accession codes can be found in the materials and methods and all computer codes are published and referenced in the text.

References

1. Shock, E. L. Hydrothermal systems as environments for the emergence of life. In Bock, G.R. and Goode, J.A. (Eds.), Evolution of Hydrothermal Ecosystems on Earth (and Mars?) Chichester, U.K. (J. Wiley & Sons Ltd.) 202, 40–60 (1996).
2. Baross, J. & Hoffman, S. Submarine hydrothermal vents and associated geographic environments as sites for the origin and evolution of life. Origins of Life and Evolution of the Biosphere 15, 327–345, https://doi.org/10.1007/BF01080177 (1985).
3. Martin, W. & Russell, M. J. On the origin of biochemistry at an alkaline hydrothermal vent. Philosophical Transactions of the Royal Society B-Biological Sciences 362, 1887–1925, https://doi.org/10.1098/rstb.2006.1881 (2007).
4. Früh-Green, G. L. Connolly, J. A. D., Plas, A., Kelley, D. S. & Grobety, B. Serpentinization of oceanic peridotites: Implications for geochemical cycles and biological activity. Subseaflow Biosphere at Mid-Ocean Ranges 144, 119–136, https://doi.org/10.1029/2004gm003104 (2004).
5. Sleep, N. H., Bird, D. K. & Pope, E. C. Serpentinite and the dawn of life. Philosophical Transactions of the Royal Society B-Biological Sciences 366, 2857–2869, https://doi.org/10.1098/rstb.2011.0129 (2011).
6. Gleen, C. R., Baross, J. A. & Waite, J. H. The pH of Enceladus’ ocean. Geochimica et Cosmochimica Acta 162, 202–219, https://doi.org/10.1016/j.gca.2015.04.017 (2015).
7. Fritz, P., Clark, I. D., Fontes, J. C., Whiticar, M. J. & Faber, E. Deutermium and 13C evidence for low-temperature production of hydrogen and methane in a highly alkaline groundwater environment in Oman. Water-Rock Interaction, Vol I: Low Temperature Environments, 793–796 (1992).
8. Shock, E. L. Stability of peptides in high-temperature aqueous-solutions. Geochimica et Cosmochimica Acta 56, 3481–3491, https://doi.org/10.1016/0012-821x(92)90322-n (1992).
9. Shock, E. L. & Schulte, M. D. Organic synthesis during fluid mixing in hydrothermal systems. Journal of Geophysical Research-Plants 103, 28513–28527, https://doi.org/10.1029/98je02142 (1998).
10. McCollom, T. M. & Seewald, J. S. Experimental constraints on the hydrothermal reactivity of organic acids and acid anions: I. Formic acid and formate. Geochimica et Cosmochimica Acta 67, 3625–3644, https://doi.org/10.1016/0016-7037(03)00136-4 (2003).
11. McCollom, T. M. & Seewald, J. S. Carbon isotope composition of organic compounds produced by abiotic synthesis under hydrothermal conditions. Earth and Planetary Science Letters 243, 74–84, https://doi.org/10.1016/j.epsl.2006.01.027 (2006).
12. Foustoukos, D. I. & Seyfried, W. E. Hydrocarbons in hydrothermal vent fluids: The role of chromium-bearing catalysts. Science 304, 1002–1005, https://doi.org/10.1126/science.1096033 (2004).
13. Proskurowski, G. et al. Abiogenic hydrocarbon production at Lost City hydrothermal field. Science 319, 604–607, https://doi.org/10.1126/science.1151194 (2008).
14. Lang, S. Q., Butterfield, D. A., Schulthe, M., Kelley, D. S. & Lilley, M. D. Elevated concentrations of formate, acetate and dissolved organic carbon found at the Lost City hydrothermal field. Geochimica et Cosmochimica Acta 74, 941–952, https://doi.org/10.1016/j.gca.2009.10.045 (2010).
15. Schrenk, M. O., Brazelton, W. J. & Lang, S. Q. In Carbon in Earth Vol. 75 Reviews in Mineralogy & Geochemistry (eds R. M. Hazen, A. Jones, & I. A. Baross) 575–606 (Mineralogical Soc Amer, 2013).
16. McCollom, T. M. Geochemoal constraints on sources of metabolic energy for chemolithoautotrophy in ultramafic-hosted deep-sea hydrothermal systems. Astrobiology 7, 933–950, https://doi.org/10.1089/ast.2006.0119 (2007).
17. Amend, J. P., McCollom, T. M., Hentscher, M. & Bach, W. Catabolic and anabolic energy for chemolithoautotrophs in deep-sea hydrothermal systems hosted in different rock types. Geochimica et Cosmochimica Acta 75, 5736–5748, https://doi.org/10.1016/j.gca.2011.07.041 (2011).
18. Reveillaud, J. et al. Subseaflow microbial communities in hydrogen-rich vent fluids from hydrothermal systems along the Mid-Cayman Rise. Environmental Microbiology 18, 1970–1987, https://doi.org/10.1111/1462-2920.13173 (2016).
19. Barnes, I., O’Neill, J.R. & Trescases, J.J. Present day serpentinization in new-caledonia, oman and yugoslavia. Geochimica et Cosmochimica Acta 42, 144–145, https://doi.org/10.1016/0016-7037(78)90225-9 (1978).
20. Neal, C. & Stanger, G. Hydrogen generation from mantle source rocks in oman. Earth and Planetary Science Letters 66, 315–320, https://doi.org/10.1016/0012-821x(83)90039-9 (1983).
21. Kelley, D. S. et al. A serpentinite-hosted ecosystem: The lost city hydrothermal field. Science 307, 1428–1434, https://doi.org/10.1126/science.1102556 (2005).
22. Cipoll, F., Gambardella, B., Marini, L., Ottonello, G. & Zuccolotti, M. V. Geochemistry of high-pH waters from serpentinites of the Gruppo di Voltri (Genova, Italy) and reaction path modeling of CO2 sequestration in serpentinite aquifers. Applied Geochemistry 19, 787–802, https://doi.org/10.1016/j.apgeochem.2003.10.007 (2004).
23. Früh-Green, G. L. et al. 30,000 years of hydrothermal activity at the Lost City vent field. Science 301, 495–498, https://doi.org/10.1126/science.1085582 (2003).
24. McCollom, T. M., Lollar, B. S., Lacrampe-Couloume, G. & Seewald, J. S. The influence of carbon source on abiotic organic synthesis and carbon isotope fractionation under hydrothermal conditions. *Geochimica et Cosmochimica Acta* 74, 2717–2740, https://doi.org/10.1016/j.gca.2010.02.008 (2010).

25. Aubrey, A., Cleaves, H. & Bada, J. The Role of Submarine Hydrothermal Systems in the Synthesis of Amino Acids. *Origins of Life and Evolution of Biospheres* 39, 91–108, https://doi.org/10.1016/s1094-0888-9153-2 (2009).

26. McDermott, J. M., Seewald, J. S., German, C. R. & Sylva, S. F. Pathways for abiotic organic synthesis at submarine hydrothermal fields. *Proceedings of the National Academy of Sciences of the United States of America* 112, 7668–7672, https://doi.org/10.1073/pnas.1506295112 (2015).

27. Schrenk, M. O., Kelley, D. S., Bolton, S. A. & Baross, J. A. Low archaeal diversity linked to subsurface geochemical processes at the Lost City Hydrothermal Field, Mid-Atlantic Ridge. *Environmental Microbiology* 6, 1086–1095, https://doi.org/10.1111/j.1462-2980.2004.00650.x (2004).

28. Brazelton, W. V. & Baross, J. Methane- and sulfur-metabolizing microbial communities dominate the Lost City hydrothermal field ecosystem. *Applied and Environmental Microbiology* 72, 6257–6270, https://doi.org/10.1128/AEM.00574-06 (2006).

29. Proskurowski, G., Lilley, M. D., Kelley, D. S. & Olson, E. J. Low temperature volatile production at the Lost City Hydrothermal Field, evidence from a hydrogen stable isotope geothermometer. *Chemical Geology* 229, 331–343, https://doi.org/10.1007/j.

30. Lang, S. Q. et al. Microbial utilization of abiogenic carbon and hydrogen in a serpentinite-hosted system. *Geochimica et Cosmochimica Acta* 92, 82–99, https://doi.org/10.1016/j.gca.2012.06.006 (2012).

31. Gerasimchuk, A. L. et al. The search for sulfate-reducing bacteria in mat samples from the lost city hydrothermal field by molecular cloning. *Microbiology* 96, 95–105, https://doi.org/10.1113/jlmso2617010153 (2010).

32. Lang, S. Q., Bernasconi, S. M. & Früh-Green, G. L. Stable isotope analysis of organic carbon in small (μg C) samples and dissolved organic acids in marine samples using wet chemical oxidation. *Rapid Communications in Mass Spectrometry* 19, 498–503, https://doi.org/10.1002/rcm.144 (2005).

33. Palandri, J. L. & Reed, M. H. Geochemical models of metasomatism in ultramafic systems: Serpentinization, rodingitization, and sea floor carbonate chimney precipitation. *Geochimica et Cosmochimica Acta* 68, 1115–1133, https://doi.org/10.1016/j.gca.2003.08.006 (2004).

34. Lowell, R. P. A fault-driven circulation model for the Lost City Hydrothermal Field. *Geophysical Research Letters* 34, L02606, https://doi.org/10.1029/2006gl028107 (2007).

35. Jansen, K., Thauer, R. K., Widdel, F. & Fuchs, G. Carbon assimilation pathways in sulfate reducing bacteria - formate, carbon-dioxide, carbon-monoxide, and acetate assimilation by desulfovibrio-baarsii. *Archives of Microbiology* 138, 257–262, https://doi.org/10.1007/bf00302132 (1984).

36. Penning, H. & Conrad, R. Carbon isotope effects associated with mixed-acid fermentation of saccharides by Clostridium papyrosolvens. *Geochimica et Cosmochimica Acta* 70, 2283–2297, https://doi.org/10.1016/j.gca.2006.01.017 (2006).

37. Bradley, A., Hayes, J. & Summons, R. Extraordinary C-13 enrichment of diether lipids at the Lost City Hydrothermal Field indicates a carbon-limit ed ecosystem. *Geochimica et Cosmochimica Acta* 73, 102–118, https://doi.org/10.1016/j.gca.2008.10.005 (2009).

38. Ferry, J. G. Formate dehydrogenase. *FEBS Microbiology Letters* 87, 377–382, https://doi.org/10.1111/j.1574-9349.1990.tb04940.x (1990).

39. Nielsen, M. B., Kjeldsen, K. U. & Ingvorsen, K. Desulfitibacter alkalitolerans gen. nov., sp nov, an anaerobic, alkalitolerant, sulfate-reducing bacterium isolated from a district heating plant. *International Journal of Systematic and Evolutionary Microbiology* 56, 2831–2836, https://doi.org/10.1099/ijs.0.04556-0 (2006).

40. Brazelton, W. V. et al. Archaea and bacteria with surprising microdiversity show shifts in dominance over 1,000-year time scales in hydrothermal chimneys. *Proceedings of the National Academy of Sciences of the United States of America* 107, 1612–1617, https://doi.org/10.1073/pnas.0905369107 (2010).

41. Pikuta, E. et al. Desulfotomaculum alkaliphilum sp nov, a new alkaliphilic, moderately thermophilic, sulfate-reducing bacterium. *International Journal of Systematic and Evolutionary Microbiology* 50, 25–33 (2000).

42. Méhay, S. et al. Record of archaeological activity at the serpentinite-hosted Lost City Hydrothermal Field. *Geology* 41, 570–592, https://doi.org/10.1111/j.1971-0106.2013.01456.x (2013).

43. Kaster, A. K. et al. More Than 200 Genes Required for Methane Formation from H-2 and CO2 and Energy Conservation Are Present in Methanothermobacter marburgensis and Methanothermobacter thermotrophicus. *Archaea-an International Microbiological Journal* https://doi.org/11055/2011/1793848 (2011).

44. Früh-Green, G., Baross, J. A. & Pecher, D. A. Mineralizing Filamentous Bacteria from the Prony Bay Hydrothermal Field Give New Insights into the Functioning of Serpentinization-Based SubseafloorEcosystems. *Frontiers in Microbiology* 8, 18, https://doi.org/10.3389/fmicb.2017.00057 (2017).

45. Weiss, M. C. et al. The physiology and habit of the last universal common ancestor. *Nature Microbiology*, 1–8 (2017).

46. Von Damm, K. L., Edmond, J. M., German, C. R. & Sylva, S. F. Pathways for abiotic organic synthesis at submarine hydrothermal fields. *Proceedings of the National Academy of Sciences of the United States of America* 112, 7668–7672, https://doi.org/10.1073/pnas.1506295112 (2015).

47. Reimer, P. J., Brown, T. A. & Reimer, R. W. Discussion: Reporting and calibration of post-bomb C-14 data. *Radiocarbon* 46, 1299–1304 (2004).
58. Boisvert, S., Raymond, F., Godzaridis, E., Laviolette, F. & Corbeil, J. Ray Meta: scalable de novo metagenome assembly and profiling. Genome Biology 13, 13, https://doi.org/10.1186/gb-2012-13-12-r122 (2012).

59. Mikheenko, A., Saveliev, V. & Gurevich, A. MetaQUAST: evaluation of metagenome assemblies. Bioinformatics 32, 1088–1090 (2016).

60. Clark, S. C., Egan, R., Frazier, P. I. & Wang, Z. ALE: a generic assembly likelihood evaluation framework for assessing the accuracy of genome and metagenome assemblies. Bioinformatics 29, (435–443 (2013).

61. Seemann, T. Prokka: rapid prokaryotic genome annotation. Bioinformatics 30, 2068–2069 (2014).

62. Hyatt, D. et al. Prodigal: prokaryotic gene recognition and translation initiation site identification. BMC Bioinformatics 11, https://doi.org/10.1186/1471-2105-11-119 (2010).

63. Ogata, H. et al. KEGG: Kyoto Encyclopedia of Genes and Genomes. Nucleic Acids Research 27, 29–34, https://doi.org/10.1093/nar/27.1.29 (1999).

64. Meyer, E. et al. The metagenomics RAST server - a public resource for the automatic phylogenetic and functional analysis of metagenomes. BMC Bioinformatics 9, 8, https://doi.org/10.1186/1471-2105-9-8 (2008).

65. Langmead, B. & Salzberg, S. L. Fast gapped-read alignment with Bowtie 2. Nature Methods 9, 357–354, https://doi.org/10.1038/nmeth.1925 (2012).

66. Edgar, R. C. MUSCLE: Multiple sequence alignment with high accuracy and high throughput. Nucleic Acid Res. 32, 1792–1797 (2004).

67. Lang, S. Q., Früh-Green, G. L., Bernasconi, S. M. & Butterfield, D. A. Sources of organic nitrogen at the serpentinite-hosted Lost City hydrothermal field. Geobiology 11, 154–169, https://doi.org/10.1111/gbi.12026 (2013).

68. Kelley, D. S. & Früh-Green, G. L. Volatile lines of descent in submarine plutonic environments: Insights from stable isotope and fluid inclusion analyses. Geochimica et Cosmochimica Acta 65, 3323–3346, https://doi.org/10.1016/S0016-7037(01)00667-6 (2001).

Acknowledgements
We thank Merle Gierga, Stewart Bishop and Madalina Jaggi (ETH-Zürich) for extensive support in the laboratory and Chris Thornton and Alex Hyer (Univ Utah) for technical support. Funding support from an InterRidge Postdoctoral Fellowship, Swiss National Science Foundation project No 200020-131922 and 200020-143891, U.S. National Science Foundation support from project No 156702 and 1536405, the NASA Astrobiology Institute Rock-Powered Life NAI, and Census for Deep Life are gratefully acknowledged. Funding for the cruise to collect samples and generate the bathymetric map was provided to D.S. Kelley, National Science Foundation grant OCE0137206.

Author Contributions
S.Q.L. designed research; S.Q.L., G.F.G., S.M.B. and W.J.B. performed research. S.Q.L., G.F.G., S.M.B., M.O.S., J.M.M. analyzed data, S.Q.L., G.F.G., S.M.B., W.J.B., M.O.S. and J.M.M. wrote the paper.

Additional Information
Supplementary information accompanies this paper at https://doi.org/10.1038/s41598-017-19002-5.

Competing Interests: The authors declare that they have no competing interests.

Publisher’s note: Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article’s Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article’s Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit http://creativecommons.org/licenses/by/4.0/.

© The Author(s) 2018