Intrinsic giant Stark effect of boron-carbon-nitride nanoribbons with zigzag edges

T. Kaneko, K. Harigaya, and H. Imamura
Nanoscale Theory Group, NRI, AIST, 1-1-1 Umezono, Tsukuba, Ibaraki 305-8568, Japan
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Electronic properties of zigzag boron-carbon-nitride (BCN) nanoribbons, where the outermost C atoms on the edges of graphene nanoribbons are replaced by B or N atoms, are theoretically studied using the first-principles calculations. We show that BCN nanoribbons are metallic, since several bands cross the Fermi level. For BCN nanoribbons in a rich H\textsubscript{2} environment, the so-called nearly free electron state appears just above the Fermi level because of the intrinsic giant Stark effect due to the internal electric field of a transverse dipole moment. The position of the nearly free electron state can be controlled by applying an electric field parallel to the dipole moment. The hydrogenation of the nitrogen atom is necessary for the appearance of the giant Stark effect in BCN nanoribbons. We also discuss the effect of stacking order on the intrinsic giant Stark effect in bilayer BCN nanoribbons.

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Graphene nanoribbon (GNR) is a nanometer-wide strip cut from a graphene (Gr) sheet on which C atoms are arranged in a honeycomb lattice. The electronic and magnetic properties of GNR have been studied intensively by several groups [1–4], which predicted the presence of a peculiar electronic state at zigzag edges, which is called the edge state. GNR can be fabricated by the use of electron-beam lithography [5], the unzipping of carbon nanotubes (CNTs) [6, 7], and the use of bottom-up processes [8]. The presence of edge states of GNR were confirmed experimentally by Tao et al. [9].

One of the most interesting properties of Gr-based nanomaterials, such as CNT and GNR, is the controllability of energy band structures using an applied electric field, i.e., the giant Stark effect [10–15]. By applying a transverse electric field, the band gap of semiconducting CNTs can be closed [10–13] and the linear dispersion of metallic CNTs around Fermi level can be bent [14, 15].

Inspired by the rich physics and functionalities of CNT and GNR, researchers have studied nanotubes and nanoribbons made of hexagonal boron-nitride (BN) sheets, which are an inorganic analogue of Gr [16]. The giant Stark effects have been predicted in BN nanotubes [17] and BN nanoribbons [18–20]. For BN nanotubes and nanoribbons, the transverse electric field shifts the lowest \(\sigma^*\)-band downward, which is often called the nearly free electron state, and the nanotubes and nanoribbons can be metallic above a critical field [17–20]. The giant Stark effect of BN nanotubes was experimentally observed by Ishigami et al. [21].

On the other hand, Ci et al. recently reported the synthesis of hybridized BN and Gr sheets using thermal catalytic chemical vapor deposition [22]. Since B and N atoms act as acceptors and donors in Gr, respectively, a nanoribbon made of the hybridized BCN sheet should have rich functionality in controlling electronic and magnetic properties. Nakamura et al., He et al., and Basheer et al. investigated zigzag GNRs with single BN edges and showed that the ribbons are metallic ferrimagnetic or half-metallic, and that such properties are realized by the coexistence of extended states and edge-localized states [23–25]. Kan et al. reported the half-metallic properties of zigzag Gr and BN hybrid nanoribbons [26].

In this letter, we theoretically study the electronic properties of the zigzag BCN nanoribbons shown in Figs. 1 (a)-(d) using the first-principles calculations. The outermost C atoms at each edge are replaced by B or N atoms. Hereafter, we call this structure a B-C-N nanoribbon. We show that B-C-N nanoribbons are metallic, since several bands cross the Fermi level. For B-C-N nanoribbons in a rich H\textsubscript{2} environment, the nearly free electron state appears just above the Fermi level due to

![FIG. 1.](color online) Schematic illustrations of B-C-N nanoribbons with \(N = 10\) for BH,NH (a), BN,N (b), B,NH (c), and B,N (d). The dotted rectangles indicate unit cells. In Panel (e), the difference in the Gibbs free energy, \(\Delta G\), is measured from the Gibbs free energy of the B,N structure as a function of the chemical potential of H\textsubscript{2} gas.
FIG. 2. (color online) The band structures of B-C-N nanoribbons with \( N = 10 \) for BH,NH (a), BN,N (b), B,NH (c), and B,N (d). The projections to \( p_z \)-orbitals of B and N atoms are respectively indicated by the red and blue circles whose radii are proportional to the magnitude of the projection.

the intrinsic giant Stark effect arising from the transverse electric dipole moment. We also show that the intrinsic giant Stark effect can be controlled by applying an electric field. We also discuss the giant Stark effect in bilayer BCN nanoribbons.

We used the projector augmented-wave method [27] and the local density approximation (LDA) [28] implemented in the Vienna \textit{ab-initio} simulation package (VASP) code [29, 30]. As shown in Figs. 2 (a)-(d), the B-C-N nanoribbon is located in the \( x-y \) plane and runs along the \( x \)-axis. Let \( N \) be the number of zigzag lines. The cell size in the \( x \)-direction was chosen to be \( a = 2.446 \) Å which is the lattice constant of free-standing Gr optimized in LDA. We imposed a vacuum region about 12 Å thick in the \( y \) and \( z \) directions. The geometry of the nanoribbons was fully optimized until the residual force fell below \( 10^{-3} \) eV/Å. The dipole correction in the \( y \)-direction is used to exclude spurious dipole interactions between periodic images [31]. The cutoff energy of the plane-wave basis was chosen to be 400 eV, and the \( k \)-points sampling was chosen to be \( 16 \times 1 \times 1 \) with a Monkhorst-Pack mesh.

First, we discuss the stability of H-terminated B-C-N nanoribbons. After the geometry is optimized, we obtained perfectly flat nanoribbons independent of H termination. The formation energy, \( E_{\text{form}} \), of a B-C-N nanoribbon at zero temperature is defined as

\[
E_{\text{form}} = E_{\text{B-C-N}} - E_{\text{BN}} - \frac{N_C}{2} E_{\text{Gr}} - \frac{N_H}{2} E_{\text{H}_2},
\]

where \( E_{\text{B-C-N}}, E_{\text{BN}}, E_{\text{Gr}}, \) and \( E_{\text{H}_2} \) are the total energies of the B-C-N nanoribbon, BN sheet, free-standing Gr, and \( \text{H}_2 \) molecules, respectively. Here \( N_C \) and \( N_H \) are the numbers of C and H atoms in the unit cell, respectively. Table 1 summarizes the formation energies of B-C-N nanoribbons with \( N = 10 \) for BH,NH, BN,N, B,NH, and B,N. We found that the BH,NH nanoribbon is the most stable in a \( \text{H}_2 \)-rich environment and that the B,N nanoribbon

\begin{table}[h]
\begin{tabular}{|c|c|c|c|}
\hline
 & BH,NH & BH,N & B,NH & B,N \\
\hline
\( E_{\text{form}} \) (eV) & 4.542 & 5.139 & 6.531 & 7.042 \\
\hline
\( d_y \) (eÅ) & 0.67 & -0.21 & 0.39 & -0.48 \\
\hline
\end{tabular}
\end{table}
is the most stable in a H₂-poor environment. The BH,N nanoribbon becomes stablest in the intermediate environment, and the B,NH nanoribbon cannot be realized under practical conditions.

Figure 2 shows the band structures of B-C-N nanoribbons with N = 10 for BH,NH (a), BH,N (b), B,NH (c), and B,N (d). The projections to pₓ-orbitals of B and N atoms are respectively indicated by the red and blue circles whose radii are proportional to the magnitude of the projection. The highest σ-band and the lowest σ*-band are indicated by the orange and green lines, respectively. The purple lines correspond to the dangling bonds of B or N atoms, which consist mainly of pₜ-orbitals.

As shown in Fig. 2 (a), the BH,NH structure is metallic, since several bands cross the Fermi level. The magnitude of projection to the pₓ-orbitals of the highest σ-band at the Γ point, which is labeled as A in Fig. 2 (a), is shown in the top Panel of Fig. 3 (a). The projections to the other orbitals, such as s and pₜ, are zero. Since the projection to the pₓ-orbitals takes a large value along the outermost zigzag line of the right side edge consisting of B atoms, the corresponding state is localized along the zigzag line, i.e., the current carrying edge state. The bottom of the lowest σ*-band, indicated by the green curve, is located just above the Fermi level. Since the effective mass of the lowest σ*-band is 1.06 m₀ with the electron mass in vacuum m₀, this band corresponds to the so-called nearly free electron state [17–20]. The nearly free electron state appears just below the Fermi level and consists of N atoms similar to that induced by the giant Stark effect in BN nanotubes and BN nanoribbons under an external electric field. However, the nearly free electron state of B-C-N nanoribbons appears in the absence of the external electric field, which implies the existence of the intrinsic giant Stark effect due to the internal electric field, i.e., the dipole moment. Actually, the hydrogenated B-C-N nanoribbons have large dipole moments in the transverse direction due to the large difference in electro-negativities among B, C, N, and H atoms. The electro-negativities of B, C, N, and H atoms are 2.04, 2.55, 3.04, and 2.20, respectively.

Calculated dipole moments in the y-direction per unit cell, dᵧ, are summarized in Table 1. The value and sign of dᵧ depend strongly on the hydrogenation pattern. The electronic density distribution of the BH,NH and BH,N structures are shown by the color plot in Figs. 3 (a) and (b), respectively. From these figures, we found that the hydrogenation of the N atom leads to the formation of a large dipole moment in the +y-direction along the N side edge of the BH,NH structure. The electric field due to the dipole moment is in the −y-direction.

Figures 3 (c) and (d) show the effects of a transverse...
electric field, $E_y$, on the band structure with $N = 10$ for the BH,NH structure. Applying the external electric field in the $+y$-direction to reduce the giant Stark effect, the nearly free electron state is shifted upward, as shown in Fig. 3 (c). In contrast, the application of the external electric field in the $-y$-direction shifts the nearly free electron states downward, as shown in Fig. 3 (d). Therefore, we conclude that the nearly free electron state in the B-C-N nanoribbon is caused by the intrinsic giant Stark effect due to the internal electric field of a dipole moment.

Let us move on to the bilayer B-C-N nanoribbons where the intrinsic giant Stark effect will depend on the stacking order. Figures 4 (a)-(c) show the top and side views of optimized structures of the bilayer B-C-N nanoribbons. We considered three different stacking orders: B-C-N/B-C-N (a), B-C-N/N-C-B (b), and N-C-B/B-C-N (c). The difference between the B-C-N/N-C-B and N-C-B/B-C-N structures is whether the outermost atom is N (b) or B (c). While we started the geometry optimization from the so-called AB-stacking structure, the AB-stacking was broken, as shown in the top view of Figs. 4 (a)-(c). The dipole moment of the B-C-N/B-C-N structure is $d_y = 0.66 \text{ eÅ}$ which induces the intrinsic giant Stark effect as shown in Fig. 4 (d). For the B-C-N/N-C-B and N-C-B/B-C-N structures, the dipole moments of the top and bottom layers cancel each other out, and the nearly free electron state around the Fermi level disappears as shown in Figs. 4 (e) and (f).

In summary, we theoretically studied the electronic properties of BCN nanoribbons with zigzag edges where the outermost C atoms are uniformly replaced by B and N atoms using the first-principles calculations. BCN nanoribbons are metallic, since several bands cross the Fermi level. For BCN nanoribbons in a H$_2$-rich environment, the nearly free electron state appears just above the Fermi level because of the intrinsic giant Stark effect due to the internal electric field of the transverse dipole moment. The hydrogenation of N atoms plays a decisive role in the appearance of the giant Stark effect in BCN nanoribbons. The strength of the intrinsic giant Stark effect can be controlled by applying an external electric field. With a decrease in the chemical potential of H$_2$ gas, the giant Stark effect disappears due to the desorption of H atoms from N atoms. We also investigated the electronic properties of bilayer BCN nanoribbons and showed that the appearance of the giant Stark effect depends on the stacking order.

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* h-imamura@aist.go.jp

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