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CHROMIUM IN SOIL ORGANIC MATTER AND COWPEA AFTER FOUR CONSECUTIVE ANNUAL APPLICATIONS OF COMPOSTED TANNERY SLUDGE

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ABSTRACT

Tannery sludge contains high concentrations of inorganic elements, such as chromium (Cr), which may lead to environmental pollution and affect human health. The behavior of Cr in organic matter fractions and in the growth of cowpea (Vigna unguiculata L.) was studied in a sandy soil after four consecutive annual applications of composted tannery sludge (CTS). Over a four-year period, CTS was applied on permanent plots (2 × 5 m) and incorporated in the soil (0-20 cm) at the rates of 0, 2.5, 5.0, 10.0, and 20.0 Mg ha⁻¹ (dry weight basis). These treatments were replicated four times in a randomized block design. In the fourth year, cowpea was planted and grown for 50 days, at which time we analyzed the Cr concentrations in the soil, in the fulvic acid, humic acid, and humin fractions, and in the leaves, pods, and grains of cowpea. Composted tannery sludge led to an increase in Cr concentration in the soil. Among the humic substances, the highest Cr concentration was found in humin. The application rates of CTS significantly increased Cr concentration in leaves and grains.

Keywords: soil contamination, humic substances, solid residue, heavy metals, plant growth, trace elements.
INTRODUCTION

The tannery industry represents an important activity in the Brazilian economy, processing 44 million units of leather annually and, consequently, generating a high quantity of tannery sludge (Santos et al., 2011). Tannery sludge contains high concentrations of inorganic elements, such as chromium (Cr), which may lead to environmental pollution and affect human health (Gupta and Sinha, 2007). However, such waste also contains organic matter and plant nutrients that can improve soil fertility and plant growth (Hargreaves et al., 2008; Santos et al., 2011). Composting tannery sludge may be useful for stabilizing nutrients presents in sludge (Castaldi et al., 2004) and decreasing sludge toxicity (Araújo and Monteiro, 2005). Some studies have already been conducted using composted tannery sludge in different crops, such as soybean and maize (Ferreira et al., 2003) and cowpea (Teixeira et al., 2006; Silva et al., 2013). These studies showed an increase in plant yield after amending the soil with composted tannery sludge. However, there were increases in Cr concentration in soils and plants treated with this residue.

The behavior of Cr in soils treated with tannery sludge is influenced by its valence state of Cr\(^{3+}\) being stable in soil (Milacic and Stupar, 1995; McBride, 2002). Usually, neutral or alkaline soils exhibit Cr in the trivalent form (Cr\(^{3+}\)), which has low solubility and mobility (Alcântara and Camargo, 2001). In addition, Cr availability to plants is influenced by soil organic matter and its different fractions (Milacic and Stupar, 1995). According to Branzini and Zubillaga (2012), organic matter is effective in immobilization of Cr in soil. We hypothesized that soil with more humified organic matter would have lower Cr bioavailability, reducing the accumulation of this metal in plants.

The aim of this study was to evaluate Cr concentration in the soil, in the organic matter fractions, and in cowpea plants after four consecutive years of application of rates of composted tannery sludge to the soil.

MATERIAL AND METHODS

The experimental site is located at the Agricultural Science Center, Teresina, Piauí, Brazil (05º 05’ S; 42º 48’ W, 75 m). Climate in the region is dry tropical (Köppen), and it is characterized by two distinct seasons, a rainy summer and dry winter, with annual average temperatures of 30 °C and annual rainfall of 1,200 mm. The rainy season extends from January to April when 90 % of total annual rainfall occurs.

Each year, composted tannery sludge (CTS) was produced by mixing tannery sludge with sugarcane bagasse and cattle manure (1:3:1; v:v:v). The composting process was carried out using the aerated-pile method for 85 days. Pile dimensions were 2 × 1 × 1.5 m (l × w × h). The pile was turned over twice a week for the first 30 days. After that, it was turned over twice a month for an additional 55 days. At the end of the composting process, 20 subsamples were randomly collected from the CTS to produce a composite sample. The N, P, and K contents were evaluated in this sample using Kjeldahl, colorimetry, and photometry methods, respectively. The other elements and trace elements (Cr, Cd, Ni, and Pb) were evaluated by atomic absorption spectrophotometry according to...
the 3050B method (USEPA, 1996), which consists of solubilization of the sample with HNO₃, HCl, and H₂O₂ concentrated under heating and absorbance reading using air-acetylene-nitrous oxide-acetylene flame (Cr) or air-acetylene flame.

For four years (2009-2012), CTS was applied on the permanent plots (2 x 5 m) and incorporated into the soil (0-20 cm) at the rates of 0, 2.5, 5.0, 10.0, and 20.0 Mg ha⁻¹ (dry weight basis). Phosphorus and K were added as necessary based on soil analysis in order to provide all the plots with the same P and K level. The P and K levels in the plots were within the range required by cowpea (Freire Filho et al., 2012). A completely randomized experimental design was used, with four replicates. Plots were marked out (20 m² each, with 12 m² useful area for soil and plant sampling), with rows spacing of 1.0 m.

In the fourth year (2012), CTS was applied ten days before sowing cowpea (Vigna unguiculata). It was spread on the soil surface and incorporated in the 20 cm layer with a harrow. Cowpea was grown at a rate of five plants m⁻¹ (about 62,000 plants ha⁻¹). Data were collected 60 days after CTS application. The shoots, pods and grains of cowpea were sampled from 10 plants in each plot.

Four soil samples were collected from each plot (0-20 cm) and pooled to form a composite sample per plot. Soil organic matter fractionation in fulvic acids, humic acids, and humin was performed according to the method proposed by the International Humic Substances Society (IHSS), as described in Swift (1996). A 0.1 mol L⁻¹ NaOH solution was used for extraction from the soil samples until exhaustion of organic matter. This extract (humic substances) was acidified to pH 1.0 with concentrated H₂SO₄, and the precipitate (humic acids) was suspended in 0.1 mol L⁻¹ NaOH solution. Chromium concentration in plant, soil, and organic matter fractions were analyzed using acid digestion with HNO₃, HCl, and H₂Cl₂ concentrated under heating and absorbance reading made in AAS using nitrous oxide-acetylene flame (USEPA, 1996). Validation of the method was carried out with certified reference material, as shown in table 1. It is important to note that the USEPA (1996) method does not estimate all the trace elements present in the soil.

Analysis of variance was used for statistical analysis and the mean values were compared by Student’s t test at 5 % significance. Regression analysis was not carried out because the mineral fertilizers used may also contain trace elements.

RESULTS AND DISCUSSION

Recovery of Cr contained in the reference analytical materials [Tomato leaves - NIST SRM 1573a, and soil contaminated with sewage sludge - RTC CRM 005-050 (USEPA, 2007)] was 92.45 % for plants and 57.09 % for soil. It is well known that the 3050B method (USEPA, 1996) is not able to totally solubilize soil, sediment, and sludge samples, so this procedure is recommended for studies on environmental contamination and pollution since it evaluates the maximal trace elements potentially available to plants (Andrade et al., 2014). The selection of this method was based on the objective of estimating the environmental availability of trace elements. Thus, recovery of the Cr in the reference samples may be considered satisfactory when compared to the methods using HNO₃, HCl, H₂O₂, and HF (Chen and Ma, 2001; Vieira et al., 2005).

Composted tannery sludge (CTS) exhibited high pH values, organic C, and Cr content (Table 2). Consequently, after four years, application of CTS significantly increased soil pH, soil organic matter (SOM), and Cr content (Table 3). Soil pH increased

| Table 1. Recovery of trace elements by the methods used in the analysis of the certified material |
|-----------------------------------------------|
| Element | Certified material(1) | Certified value | Recovery |
|--------|----------------------|----------------|----------|
|        |                      | mg kg⁻¹         | %        |
| Cd     | NIST SRM 1573a       | 1.52            | 1.87     | 123.0 |
|        | RTC CRM 005-050      | 13.70           | 13.76    | 100.4 |
| Cr     | NIST SRM 1573a       | 1.99            | 1.04     | 55.3  |
|        | RTC CRM 005-050      | 41.30           | 48.06    | 117.2 |
| Cu     | NIST SRM 1573a       | 4.70            | 4.46     | 94.9  |
|        | RTC CRM 005-050      | 465.40          | 387.75   | 83.3  |
| Ni     | NIST SRM 1573a       | 1.59            | 0.42     | 26.4  |
|        | RTC CRM 005-050      | 26.00           | 29.95    | 115.2 |
| Zn     | NIST SRM 1573a       | 30.90           | 29.43    | 95.2  |
|        | RTC CRM 005-050      | 625.20          | 619.25   | 99.0  |

(1) Tomato leaves - NIST SRM 1573a and soil contaminated with sewage sludge - RTC CRM 005-050.
above 7.0 due to CTS rates and, at alkaline pH values, heavy metals remain inert in the soil under forms with low mobility (Hayes and Traina, 1998). At pH values above 5.0, especially Cr is in an insoluble form \([\text{Cr(OH)}_3]\) (Aquino Neto and Camargo, 2000), reducing its toxic potential. Although Cr concentration increased in the soil, the values found are below the limit for agricultural soils (75 mg kg\(^{-1}\)) according to Conama (2009). Even at the highest CTS rate, Cr concentration was half the limit for soil. However, from the first to the fourth year, Cr concentration in the soil showed more than a fivefold increase.

Results for Cr concentration in the organic matter fractions showed that more than 57 % of the metal present in the soil was bound to humic substances, and the highest concentration was found in the humin (Table 3).

Humic substances contain a large number of complexing sites; hence they behave as a natural “multiligand” complexing system (Buffle, 1988). The high degree of selectivity of SOM for most trace elements in cationic form indicates that they form inner-sphere complexes with the functional groups, often forming an internal five- or six-member ring on structures (Huang and Germida, 2002; Sparks, 2003). The high Cr bond with humin occurred because this fraction is more stable and strong for complex mineral elements. The Cr concentrations in the humic and fulvic acids were very similar (Table 3). Unlike Cr bound to humin, Cr bound to humic and fulvic acids is more available to plants and to percolation through the soil profile. Composted tannery sludge increased Cr concentration in the organic fractions, fulvic acids, humic acids, and humin, compared to the control. Cr concentration in the fulvic acid and humin fractions increased with the increase in the rate of compost application, but this was not observed for the humic acid fraction. This result was probably due to the time between compost applications and sampling, which was enough for transformation of the humic acids to humin, but not for polymerization of fulvic acids to humic acids; or the velocity of the transformation of fulvic acids to humic acids was higher than the transformation of fulvic acids to humic acids. Finally, it is possible that the transformation of the composted organic matter to fulvic acids was higher than the transformation of fulvic acids to humic acids and, consequently, to humin.

Chromium concentration significantly increased in leaves and grains after CTS application (Table 3). There was not a significant difference in Cr concentration in the pods. The increase in Cr concentration in the leaves may be related to higher soil Cr concentration after CTS application and, therefore, the metal was translocated from the roots to the leaves. This difference in Cr accumulation in

### Table 2. Chemical properties of composted tannery sludge (CTS) and the limits permitted for heavy metal (Limit) by National Environment Council (Conama)

| Property | Composted tannery sludge | Limit(1) |
|----------|---------------------------|----------|
|          | 2009 | 2010 | 2011 |
| pH       | 7.8  | 7.2  | 7.5  |
| C (g kg\(^{-1}\)) | 187.5 | 195.3 | 201.2 |
| N (g kg\(^{-1}\)) | 1.28 | 1.39 | 1.51 |
| P (g kg\(^{-1}\)) | 4.02 | 3.83 | 4.91 |
| K (g kg\(^{-1}\)) | 3.25 | 3.51 | 2.90 |
| Ca (g kg\(^{-1}\)) | 95.33 | 84.28 | 121.18 |
| Mg (g kg\(^{-1}\)) | 6.80 | 5.71 | 7.21 |
| S (g kg\(^{-1}\)) | 9.39 | 8.43 | 10.20 |
| Cu (mg kg\(^{-1}\)) | 17.80 | 19.51 | 16.38 |
| Zn (mg kg\(^{-1}\)) | 141.67 | 128.31 | 127.81 |
| Ni (mg kg\(^{-1}\)) | 21.92 | 28.61 | 23.26 |
| Cd (mg kg\(^{-1}\)) | 2.87 | 3.93 | 1.93 |
| Cr (mg kg\(^{-1}\)) | 2.26 | 2.58 | 1.94 |
| Pb (mg kg\(^{-1}\)) | 42.67 | 38.54 | 40.31 |

(1) According to Conama (2009).

### Table 3. pH values, soil organic matter (SOM), and Cr in the soil, fulvic acids (Cr-FA), humic acids (Cr-HA), and humin (Cr-HUM) fractions and chromium concentration in leaves, pods, and grains of cowpea after four years of composted tannery sludge (CTS) application

| CTS Mg ha\(^{-1}\) | pH | SOM g kg\(^{-1}\) | Cr-soil mg kg\(^{-1}\) | Cr-FA mg kg\(^{-1}\) | Cr-HA mg kg\(^{-1}\) | Cr-HUM mg kg\(^{-1}\) | Cr-leaf mg kg\(^{-1}\) | Cr-pod mg kg\(^{-1}\) | Cr-grain mg kg\(^{-1}\) |
|---------------------|----|-----------------|----------------------|---------------------|---------------------|----------------------|----------------------|----------------------|----------------------|
| 0                   | 6.6 b | 17.88 c | 6.73 e | 0.63 c | 0.71 b | 2.5 d | 1.56 d | 1.24 a | 0.55 c |
| 2.5                 | 6.9 b | 19.12 c | 12.7 d | 0.86 c | 2.12 a | 7.9 c | 3.00 c | 1.04 a | 0.99 b |
| 5.0                 | 7.3 a | 23.41 b | 26.8 c | 1.32 b | 1.65 a | 13.4 b | 5.41 b | 0.99 a | 1.11 b |
| 10.0                | 7.7 a | 25.34 b | 30.4 b | 1.29 b | 1.96 a | 18.2 a | 7.80 a | 1.15 a | 1.62 a |
| 20.0                | 7.9 a | 28.18 a | 36.9 a | 2.25 a | 2.29 a | 19.1 a | 7.68 a | 1.08 a | 1.74 a |

Values followed by the same letter within each column are not significantly different at the 5 % as determined by Student’s \(t\)-test.
the different parts of the plant suggests different cellular mechanisms of bioaccumulation of Cr and this may control Cr translocation and partitioning in the plant. Therefore, more Cr was accumulated in the leaves and grains than in the pods, and as Cr is a non-essential element, the plants may not possess any specific mechanism for transporting the Cr (Kumar and Chopra, 2012).

Although Cr concentration in the leaves increased, it did not exceed the limits proposed by Macnicol and Beckett (1985) for the bean crop (10 mg kg⁻¹). However, Cr concentration in grain was above the limit (0.1 mg kg⁻¹) for food and grains established by Anvisa (1965) and the limit mentioned by Kabata-Pendias and Pendias (2001) (0.15 mg kg⁻¹).

**CONCLUSIONS**

The application rates of composted tannery sludge applied to soils for four years led to an increase in the Cr levels in soil and fractions of the soil organic matter and tended to stabilize at the rates of 10.0 and 20.0 Mg ha⁻¹.

The highest Cr concentration in the soil organic matter fractions is that bound to humin.

Soil Cr is available to be taken up by cowpea and it accumulates in leaves and grains.

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