Comparative numerical study of single and two-phase models of nanofluid liquid film evaporation in a vertical channel

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Abstract. The main purpose of this study is to survey numerically comparison of two-phase and single-phase models of heat and mass transfer of Al2O3-water nanofluid liquid film flowing downward a vertical channel. A finite difference method is developed to produce the computational predictions for heat and mass transfer during the evaporation of the liquid film approached by the single-phase and two-phase models. The model solves the coupled governing equations in both nanofluid and gas phases together with the boundary and interfacial conditions. The systems of equations obtained by using an implicit finite difference method are solved by Tridiagonal Matrix Algorithm. The results show that the two-phase model is more realistic since it takes into account the thermophoresis and Brownian effects.

1 Introduction

The enhancement of heat transfer efficiency has been for decades the mean objective for heat exchangers manufacturer and research laboratories. Therefore, many active and passive solutions are being proposed. In this context, flowing liquid films have the advantage of higher heat transfer coefficient in comparison to single-phase flows. Flowing liquid films are largely described in the literature. Early Yan et al. [1] have experimentally analysed the evaporative cooling of a liquid film in a vertical channel with insulated walls. The results show a better cooling of the liquid film when low film flow rate and high inlet temperature were settled. The combined buoyancy effects on mass and heat transfer were examined by Feddaoui et al. [2]. They reported in their paper the best conditions to foster the heat transfer. Nait alla et al. [3, 4] investigated deeply the evaporation of alcohols and glycols inside fully and partially heated channels. Heated zones number and inlet conditions effects on the heat and mass transfer were discussed. Recently, Najim et al. [5] analysed the effect of salinity on the evaporation and heat exchange of a falling film inside a vertical tube. The authors reported the convenient boundary conditions for a better evaporation in desalination applications.

Nowadays, nanofluids seem to be a good passive solution. Plenty of problems are already investigated numerically as well as experimentally. However, those papers deal with problems containing one flowing fluid, and few papers devoted to phase change problems are restricted to droplets and motionless liquid films, see for instance [6, 7]. The numerical examination of nanofluids is done either by the homogeneous model or by the two-phase model. For example, Mounmi et al. [8] studied the nanofluid effect on the mixed convection in lid-driven cavity. The used model assumes a uniform distribution of nanoparticles. Avramenko et al. [9] studied the nanoparticles effect in motionless film boiling. The analytical results showed that the increase of nanoparticles concentration promotes heat and mass transfer. Najim et al. [10] studied the effect of nanoparticles concentration on the heat and mass transfer during the evaporation of nanofluids inside a vertical channel. The authors claimed that nanoparticles presence improves the evaporation rate of the liquid film.

The main aim of this study is to predict heat and mass transfer characteristics by single phase (homogeneous fluid) and two-phase models in order to compare the two models. We are not aware of any published study on nanofluids effect on falling liquid film evaporation especially by the two-phase model excluding the study of Najim et al. [10]. The problem description and mathematical formulation are specified below.

2 Mathematical modelling

2.1 Physical problem
Since the two flows take place in a downward direction the fields of velocity, pressure, temperature and concentrations at each section depend only on the upstream solutions. This property gives the parabolic character to the governing equations. Moreover, the difference between the wall and the liquid film interface temperatures is small. In consequence, the vapour diffusion is done at very small velocities, which allows neglecting the interface shear stress. For Reynolds number less than 1500, the liquid film flow may be considered laminar as quoted by Ueda and Tanaka [11].

![Flow Diagram](image)

**Fig. 1.** The geometry of the channel with boundary conditions.

### 2.2 Governing equations

Proceeding with the above assumptions, the governing equations for laminar steady state flows are the continuity, momentum, energy species concentration and nanoparticles concentration equations. For the liquid phase those equations are:

\[
\nabla \cdot (\rho_n \vec{v}_n) = 0
\]

\[
\nabla \cdot (\rho_n \vec{v}_n \cdot \nabla) = -\nabla P + \nabla \left( \mu_n \nabla \vec{v}_n \right) + \rho_n g
\]

\[
= \nabla \cdot \left( \lambda_n \nabla T_n \right)
\]

\[
+ C_p \rho_n \nabla \left( D_g \phi \nabla T_n + \frac{\nabla^2 T_n}{T_n} \right)
\]

\[
\nabla \cdot (\rho_p \phi \vec{v}) = \nabla \cdot \left( \rho_p D_B \nabla \phi + \rho_p D_T \frac{\nabla T_n}{T_n} \right)
\]

The gas flow governing equations are similar to the liquid film, thus, with the presence of the air, the gas flow is ternary and the equations may be written as:

\[
\nabla \cdot (\rho_g \vec{v}_g) = 0
\]

\[
\nabla \cdot (\rho_g \vec{v}_g \cdot \nabla) = -\nabla P + \nabla \left( \mu_g \nabla \vec{v}_g \right) + \rho_g g
\]

2.3 Boundary conditions

Boundary conditions are associated with the previous equations to solve the mathematical problem. Uniform velocity, temperature, mass fractions and nanoparticles concentration are specified at the channel inlet as follows:

- In liquid phase:
  \[\Gamma = \Gamma_e; \quad T = T_{in}\]

- In gas phase:
  \[u = u_0; \quad T = T_0; \quad P = P_0; \quad w = w_0\]

- At the wet wall (y = d):
  \[\frac{\partial T}{\partial y} = q_w; \quad u = 0; \quad D_g \frac{\partial \phi}{\partial y} = D_t \frac{\partial T}{\partial y}\]

- At the dry wall (y = 0):
  \[\lambda \frac{\partial T}{\partial y} = 0; \quad u = 0; \quad \frac{\partial w}{\partial y} = 0\]

The matching conditions at the interface (y = d − δx) are:

\[u_l (x) = u_{in}; \quad T_l (x) = T_{in}\]

\[\tau_l = \left[ \frac{\mu}{\partial y} \right]_{nf} = \left[ \frac{\mu}{\partial y} \right]_{G}\]

\[\frac{\partial T}{\partial y} = \left[ \frac{\partial T}{\partial y} \right]_{nf} = \left[ \frac{\partial T}{\partial y} \right]_{G}\]

Where \(h_{fg}\) is the latent heat of vaporization. The evaporating mass flux is:

\[m = -\rho_g D \frac{\partial w}{\partial y}\]

\[D_h \frac{\partial \phi}{\partial y} = D_t \frac{\partial T}{\partial y}\]

2.4 Nanofluids properties

In this paper, we investigate the nanofluid of water containing the nanoparticles of Alumina Al₂O₃. To calculate the thermophysical properties of nanofluids the following equations are used:

\[\rho_{nf} = \phi \rho_p + (1 - \phi) \rho_L\]

\[(\rho CP)_{nf} = \phi (\rho CP)_p + (1 - \phi) (\rho CP)_L\]

\[\lambda = \lambda_L(1 + 7.47\phi)\]

\[\mu_{nf} = \mu_L(1 + 39.11\phi + 533.9\phi^2)\]

2.5 Solution method

To solve governing equations, a fully implicit scheme is employed. The numerical solution is realized using finite
difference method. Each finite-difference equation system forms a tridiagonal matrix, which can be solved using the TDMA method (Patankar [12]). It is still necessary to satisfy the global mass flow constraint. This is done by correcting the pressure gradient and axial velocity profile at each axial step, according to Raithby and Schneider method [13]. Moreover, the generated grid is non-uniform, in order to enhance numerical solutions accuracy.

3 Results and discussion

After the discretization of the governing equations combined with boundary conditions, calculations are performed for water-Al2O3 nanofluid. The aim of this study is to compare the homogeneous model with the two-phase model for computing the effect of nanoparticles on the heat and mass transfer. Details about the homogenous model applied to liquid film evaporation are given in [10]. To compare the two models several runs have been done for different conditions. The parameters of each run are described in the table below for 2 meters high and 2 cm wide channel and for a gas stream flowing at Re=2000.

Table 1. Inlet and boundary conditions of each simulation run.

| Run   | \( \Gamma_0 \) (kg/m·s) | \( T_{\text{L0}} \) (°C) | \( q_W \) (W/m²) | \( \phi_0 \) (%) |
|-------|--------------------------|--------------------------|------------------|------------------|
| Run 1 | 0.01                     | 25                       | 2000             | 2                |
| Run 2 | 0.02                     | 25                       | 2000             | 2                |
| Run 3 | 0.01                     | 50                       | 2000             | 2                |
| Run 4 | 0.01                     | 25                       | 4000             | 2                |
| Run 5 | 0.01                     | 25                       | 2000             | 4                |

![Fig. 2. Latent heat flux evolution along the channel.](image)

Figure 3 demonstrates the sensible heat flux evolution along the channel as predicted by the two models. The sensible heat flux increases near the inlet to \( x=0.26 \)m for run 1 and 5 and to \( x=0.18 \)m for run 4, and then decreases continuously to the channel outlet. For the highest inlet liquid film flow rate, the sensible heat flux increases along the channel while for the highest liquid film temperature \( q_S \) decreases. It is obvious that the sensible heat flux predictions by the single-phase model are initially under those of the two-phase model for a given distance from the inlet, afterward they become above the two-phase predictions. This distance is decreased from 0.65m to 0.25m by the increase of the heat flux at the wall. However, for the run 2 where the liquid film mass flow rate is doubled, the sensible heat flux predicted by two-phase model is always larger than the single-phase predictions. At the channel outlet, the difference between the two predictions is very small except for run 2. In fact, the homogeneous fluid hypothesis assumes that both base fluid and nanoparticles have the same properties in a given section. In reality, the nanoparticles do not keep a uniform distribution in the base fluid due to the thermophoresis and Brownian motions. In consequence, doubling the heat flux applied to the wall intensifies the thermophoretic force and the nanoparticles migrate from the hot zone (wall) to the colder zone (liquid film interface). In contrast, at high liquid film temperature (run 3) the liquid film approaches the saturation state, thus, the majority of the applied heat flux at wall is converted to latent heat flux. For this reason, the homogenous and two-phase models predictions do not present a significant difference for high liquid film temperature. The inversion of the two models predictions order is due to the migration of nanoparticles. In fact, as the nanoparticles migrate to the interface, their concentration decreases near the hot zone. This leads to the decrease of the thermal conductivity near the wall. Consequently, less heat flux is transferred through the liquid film. This feature is not considered in the homogeneous model.
Fig. 3. Sensible heat flux distribution along the channel.

Figure 4 shows the evolution of accumulated evaporation rate along the channel. The results confirm the last findings and show that \( MR \) increases along the channel. The evaporation is enhanced by doubling the wall heat flux and by increasing the liquid film temperature at the inlet. In contrast, the elevation of the liquid film mass flow rate decreases considerably the evaporation rate in comparison to the reference case. The accumulated evaporation rate evolution is very similar to the latent heat flux. Thus, higher difference between the two predictions is observed for high wall heat flux (run 4) while a very small difference is observed for high liquid film temperature. Those differences are due to the nanoparticles migration form the hot zone near the wall to the colder interface as already explained.

Fig. 4. Accumulated evaporation rate evolution along the channel.

Figures 5, 6, and 7, show the interfacial, gas bulk temperature and wall temperature along the channel. The plots indicate that those temperatures evolutions are very similar to the latent heat flux. As the liquid film flows down the channel, it absorbs the heat from the wall and its interfacial temperature increases. As the interface reaches the saturation condition, the evaporation begins and the gas flow absorbs the heat flux from the liquid film by the latent and sensible heat transfer modes. This leads to the increase of the gas flow temperature along the channel. Again, all values predicted by the single-phase model are underestimated and the lowest difference occurs for high inlet liquid film temperature, while the biggest difference is observed for higher wall heat flux and liquid film temperature.

Fig. 5. Evolution of the interface temperature along the channel.

Fig. 6. Gas bulk temperature variation along the channel.

Fig. 7. Wall temperature evolution along the channel.
4 Conclusions

The liquid film evaporation inside a vertical channel with suspended $\text{Al}_2\text{O}_3$ nanoparticles is investigated numerically. The single phase (homogeneous fluid) and two-phase (base fluid+nanoparticles) models are used to run the simulations. Given that two-phase model considers the thermophoresis and Brownian motion, the interaction between base fluid and nanoparticles is computed and more realistic results were obtained. The results showed the existence of a difference between the two models predictions. However, for high liquid film temperature the difference is very small. The largest difference between the two models is observed for high wall heat flux and inlet liquid film flow rate. The single-phase model under-predict the results for all plotted parameters except for the sensible heat flux.

Nomenclature

\[ C_p \quad \text{Specific heat, J.kg}^{-1}.\text{K}^{-1} \]
\[ d \quad \text{Channel width, m} \]
\[ D \quad \text{Mass diffusivity, m}^2.\text{s}^{-1} \]
\[ D_b \quad \text{Brownian diffusion coefficient} \]
\[ D_T \quad \text{Thermophoretic diffusion coefficient} \]
\[ \dot{m} \quad \text{Evaporation mass flux, kg.s}^{-1}.\text{m}^{-2} \]
\[ M_r \quad \text{Mass Evaporation rate} \]
\[ P \quad \text{Pressure, Pa} \]
\[ q_L \quad \text{Latent heat flux, W.m}^{-2} \]
\[ q_S \quad \text{Sensible heat flux, W.m}^{-2} \]
\[ T \quad \text{Temperature, K} \]
\[ u \quad \text{Velocity component in x direction, m.s}^{-1} \]
\[ v \quad \text{Velocity component in y direction, m.s}^{-1} \]
\[ w \quad \text{Vapour mass fraction} \]
\[ x \quad \text{Longitudinal coordinate} \]
\[ y \quad \text{Transversal coordinate} \]

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