Bioactivity-Guided Separation of an \( \alpha \)-Amylase Inhibitor Flavonoid from \textit{Salvia virgata}

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Abstract

It is now believed that the inhibition of carbohydrate hydrolyzing enzymes (CHEs) in the digestive tract can significantly prolong the overall carbohydrate digestion time and decrease the postprandial hyperglycemia after a meal. Therefore, inhibitors of CHEs can be useful therapeutic approaches in the management of diabetes mellitus, especially in the type 2, and complications associated with the disease. In our previous study, the ethanol extract of the aerial parts of \textit{Salvia virgata} showed an inhibitory effect on pancreatic \( \alpha \)-amylase \textit{in-vitro}. Bioassay-guided fractionation of the extract using the \( \alpha \)-amylase inhibitory assay led to the isolation and identification of an active flavone compound, chrysoeriol. The compound concentration dependently inhibited the \( \alpha \)-amylase activity with an IC\(_{50}\) value of 1.27 (1.21-1.33) mM.

Keywords: \( \alpha \)-Amylase inhibitor; Diabetes mellitus; Chrysoeriol; \textit{Salvia virgata}.

Introduction

Diabetes mellitus is an endocrinial chronic disease characterized by elevated blood glucose levels (1-3). The control of hyperglycemia is critical in the management of diabetes mellitus since in long term, acute and chronic complications can occur (1, 4, 5). One goal of therapy for diabetic patients, especially non-insulin-dependent diabetes mellitus (type 2 diabetes), is the maintenance of normal blood glucose levels after a meal (postprandial hyperglycemia) (3, 6). A therapeutic approach for decreasing postprandial hyperglycemia is to retard and reduce the digestion and absorption of ingested carbohydrates by the inhibition of carbohydrate-hydrolyzing enzymes, such as \( \alpha \)-amylase and/or \( \alpha \)-glucosidases, in the digestive organs (1, 5, 7, 8). Therefore, there is a need to develop compounds with enzyme inhibitory activities, for which the medicinal plants may serve as potential sources (9, 10).

The use of natural products as complementary approaches in existing medications for the treatment of diabetes mellitus is growing worldwide and many plants in different countries are known to have antidiabetic effects (11). Grover \textit{et al.} reported that more than 1100 plant species have been used ethnopharmacologically or experimentally to treat diabetes mellitus (12). \textit{Salvia} is one of the largest geniuses in Labiatae family. It comprises nearly 900 species throughout the world and 58 species in Iran (13, 14). Different species of \textit{Salvia} have a long history of use for medical purposes in many countries (15, 16). On the other hand, investigations conducted on various species of \textit{Salvia} show that the plants have wide and diverse biological activities especially antioxidant, anti-inflammatory, spasmolytic, antidiabetic,


Experimental

Plant material
The aerial parts of Salvia virgata Jacq. (Synonym: Salvia sibthorpii Sibth. and Sm.) were collected from Tehran province during the flowering period in summer 2006. Voucher specimens were deposited at the Herbarium of the School of Pharmacy, Shahid Beheshti University of Medical Sciences, Tehran, Iran. The plant was dried at ambient temperature with active ventilation.

Chemicals
All of the chemical reagents used in this study were purchased from Sigma-Aldrich Chemical Co. (France) and/or Merck Company (Germany). Acarbose was obtained from Quimica Farmaceutica Bayer, S.A. (Barcelona).

Extraction, chromatography and spectroscopy
The dried and ground plant (250 g) was extracted with ethanol 90% three times by maceration method and then, the extract was concentrated in vacuo. The crude extract (12 g) was diluted with water and partitioned with n-C$_6$H$_{12}$, CHCl$_3$ and EtOAc, successively. The inhibitory effects of the crude extract and all fractions were studied on α-amylase activity. The ethyl acetate fraction displayed the highest inhibitory activity. The fraction was subjected to more fractionation by column chromatography on silica gel using chloroform/ethyl acetate system as the eluent. The polarity of the eluent was increased by increasing the ratio of EtOAc during the process. Fraction F10 which showed a high inhibitory activity, was purified by repeated preparative layer chromatography on coated plates with silica gel (230-400 mesh) using CHCl$_3$/EtOAc/HCOOH (45:45:10, v/v/v) as the best developing solvent system. Finally, fraction F10 yielded a pure active compound (53 mg).

α-Amylase inhibition test
The α-amylase inhibitory activity was determined using the method described previously by Nickavar et al. (22). Briefly, 1 mL of the porcine pancreatic α-amylase enzyme solution (0.5 IU/mL) in 20 mM phosphate buffer (pH 6.9) was incubated with 1 mL of each test (at various concentrations) for 30 min. The reaction was initiated by adding 1 mL of 0.5% soluble potato starch solution and the mixture was incubated for 3 min at 25°C. Then, 1 mL of the color reagent (96 mM 3,5-dinitrosalicylic acid and 5.31 M sodium potassium tartrate in 2 M sodium hydroxide) was added and the mixture was placed in a water bath at 85°C. After 15 min, the reaction mixture was diluted with distilled water and the absorbance value was determined at 540 nm. Individual blanks were prepared for correcting the background absorbance. In this case, the color reagent solution was added prior to the addition of starch solution and the mixture was then placed in the water bath immediately. Controls were representative of the 100% enzyme activity. They were conducted in an identical fashion replacing tests with 1 mL of the solvent.
Acarbose, a well-known \( \alpha \)-amylase inhibitor, was used as positive control. The inhibition percentage of \( \alpha \)-amylase was assessed by the formulae (1):

\[
I_{\alpha \text{-amylase}}(\%) = 100 \cdot \left( \frac{A_{\text{control}} - A_{\text{sample}}}{A_{\text{control}}} \right)
\]

Here, \( A_{\text{control}} \) is the absorbance of each control and \( A_{\text{sample}} \) is the net absorbance of each sample. The net absorbance of each sample was calculated by the equation (2):

\[
A_{\text{sample}} = A_{\text{test}} - A_{\text{blank}}
\]

In this equation, \( A_{\text{test}} \) is the absorbance of each test and \( A_{\text{blank}} \) is the absorbance of each blank.

The \( I_{\alpha \text{-amylase}}(\%) \) for each sample was plotted against the logarithm of the sample concentration, and a logarithmic regression curve was established in order to calculate the IC\(50\) value.

**Results and Discussion**

The ethanol extract of \( S. \) virgata showed a dose-dependent inhibitory effect on the \( \alpha \)-amylase activity [IC\(50\) \(= 19.08 \text{ (18.61-19.56)} \text{ mg/mL}\)] (Table 1).

In order to identify the active components, solvent-solvent partition performed with \( n-C_6H_{12}, \text{CHCl}_3 \) and EtOAc, successively. The ethyl acetate fraction revealed the highest activity so it was selected for further separation. The chromatographical analysis of the ethyl acetate fraction showed flavonoid compounds. The most active flavonoid compound was isolated as the pale yellow amorphous powder (53 mg). It had \( \text{ca} R_f = 0.7 \) on TLC (silica gel 60) with CHCl\(_3\)/EtOAc/HCOOH (45:45:10, v/v/v). The spectroscopic data for the compound were as follows:

\( ^{1}H \)-NMR (400 MHz, in DMSO-d\(_6\)), \( \delta \) : 3.93 (3H, s, OCH\(_3\)-3'), 6.18 (1H, br. s, H-6), 6.45 (1H, br. s, H-8), 6.58 (1H, s, H-3), 6.89 (1H, d, \( J = 8 \text{ Hz, H-5'} \)), 7.37 (1H, br. s, H-2'), 7.40 (1H, d, \( J = 8 \text{ Hz, H-6'} \)).

\( ^{13}C \)-NMR (100 MHz, in DMSO-d\(_6\)), \( \delta \) : 54.8 (OCH\(_3\)), 93.2 (C-8), 98.1 (C-6), 102.1 (C-3), 102.9 (C-10), 110.5 (C-2'), 115.7 (C-5'), 120.6 (C-6'), 130.4 (C-1'), 146.1 (C-3'), 148.8 (C-4'), 154.4 (C-2), 155.7 (C-9), 161.3 (C-5), 167.6 (C-7), 171.0 (C-4).

EI-MS (70 eV), \( m/z \) (I\%): 300 (10\%), 286 (82\%), 153 (26\%), 151 (20\%).

The spectral data of the compound showed that it was chrysoeriol (Figure 1) and all of its data were matched with those reported in the literature (24, 25).

In this study, chrysoeriol inhibited \( \alpha \)-amylase activity in a dose-dependent manner (Figure 2). Each point represents the mean of five experiments and the vertical bars represent the SEM. The graphs were plotted using the computer software GraphPad Prism 3.02 for Windows.
activity in a dose-dependent manner. The IC_{50} values for \(\alpha\)-amylase inhibition by chrysoeriol and acarbose (as the positive control) were 1.27 (1.21-1.33) mM and 0.049 (0.042-0.056) mM, respectively (Figure 2 and Table 1).

The genus *Salvia* generally produces a variety of phenolic metabolites, especially flavonoids, which have received much attention due to their relevant biological properties (13). Phytochemical literature survey on *S. virgata* shows the occurrence of few hydroxycinnamic acid derivatives (such as rosmarinic acid, caffeic acid, *etc.* ) and flavonoids (such as salvigenin, luteolin and its glycosides, luteolin 7,3’ ,4’-trimethyl ether, *etc.* ) (13, 26, 27). On the other hand, chrysoeriol has already been isolated from few *Salvia* species including *S. candidissima, S. dorrii, S. lavandulaefolia, S. mizrayana*, and *S. palaestina* (13). However, to the best of our knowledge, this is the first report on the isolation and identification of chrysoeriol from *S. virgata* and the inhibitory effect of the compound on \(\alpha\)-amylase activity.

**Table 1.** \(\alpha\)-Amylase inhibitory activities and IC_{50} values of the aerial parts of *S. virgata* and its active compound chrysoeriol.

| Concentration (mg/mL) | Inhibition (%) | IC_{50}a |
|-----------------------|---------------|---------|
| 36.00                 | 83.69 ± 1.15  |         |
| 28.80                 | 74.28 ± 0.49  |         |
| 23.04                 | 70.92 ± 0.74  | 19.08 (18.61-19.56) mg/mL |
| 18.43                 | 30.83 ± 1.18  |         |
| 14.75                 | 18.37 ± 0.61  |         |
| Chrysoeriol (mM)      |               |         |
| 3.48                  | 96.28 ± 1.38  |         |
| 2.23                  | 70.62 ± 0.95  |         |
| 1.42                  | 58.98 ± 1.20  | 1.27 (1.21-1.33) mM |
| 0.91                  | 28.05 ± 0.80  |         |
| 0.58                  | 11.91 ± 0.85  |         |

* The data are expressed as mean ± SEM for five experiments in each group. *a* The IC_{50} values were established by logarithmic regression curves with normalized data (using the computer software GraphPad Prism 3.02 for Windows) and presented as their respective 95% confidence limits.

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