Cytotoxic and genotoxic potential of food-borne nitriles in a liver in vitro model

Franziska Kupke1,*, Corinna Herz2,##,*, Franziska S. Hanschen3, Stefanie Platz1, Grace A. Odongo1,2,##, Simone Helmig4, Maria M. Bartolomé Rodríguez5, Monika Schreiner3, Sascha Rohn1 & Evelyn Lamy2

Isothiocyanates are the most intensively studied breakdown products of glucosinolates from Brassica plants and well recognized for their pleiotropic effects against cancer but also for their genotoxic potential. However, knowledge about the bioactivity of glucosinolate-borne nitriles in foods is very poor. As determined by GC-MS, broccoli glucosinolates mainly degrade to nitriles as breakdown products. The cytotoxicity of nitriles in human HepG2 cells and primary murine hepatocytes was marginal as compared to isothiocyanates. Toxicity of nitriles was not enhanced in CYP2E1-overexpressing HepG2 cells. In contrast, the genotoxic potential of nitriles was found to be comparable to isothiocyanates. DNA damage was persistent over a certain time period and CYP2E1-overexpression further increased the genotoxic potential of the nitriles. Based on actual in vitro data, no indications are given that food-borne nitriles could be relevant for cancer prevention, but could pose a certain genotoxic risk under conditions relevant for food consumption.

Results from human epidemiologic studies point towards a positive correlation between cancer prevention of many different organs and consumption of Brassica vegetables1. Yet, knowledge is still far from a complete understanding of the combinatory effects of secondary plant metabolites present in these vegetables as evident by conflicting results on the associations of Brassica consumption with cancer. One group of secondary plant metabolites being made responsible for the observed protective effect against cancer are glucosinolates (GSL) or more precisely their breakdown products in the form of isothiocyanates (ITC)2. The stability of GSL is comparatively low and they are easily hydrolyzed either by the plant-endogenous enzyme myrosinase after disruption of the plant tissue3, or by human microbiota-deriving myrosinase-like activities4, or degraded thermally-induced (e.g. during food processing)5,6. From the GSL hydrolysis products, ITC have been studied most intensively. By now, they are well known for their pleiotropic mechanisms of interference with carcinogenesis2,7,8. However, enzymatic degradation not only leads to the formation of ITC, but can also lead to the formation of nitriles as main products: Especially, when plants contain modifying proteins such as the epithiospecifier protein or the nitrile-specifier proteins, enzymatic degradation of GSL can be altered in favor of nitriles9,10. Moreover, these compounds can be formed also by a non-enzymatic pathway: Several studies showed that thermally-induced degradation, e.g., during the cooking of vegetables, can result in the predominant formation of nitriles being chemically more stable than ITC5,11. Whether nitriles are also relevant nutritional factors in terms of cancer prevention by consumption of Brassica vegetables has been barely addressed, to date. In order to evaluate this, better knowledge on their bioactivity, but also on their concentrations in processed vegetables is indispensable. So far, some research

1University of Hamburg, Hamburg School of Food Science, Institute of Food Chemistry, Grindelallee 117, 20146 Hamburg, Germany. 2University of Freiburg–Medical Center, Institute of Environmental Health Sciences, Molecular Preventive Medicine, Breisacher Strasse 115b, 79106 Freiburg, Germany. 3Leibniz Institute of Vegetable and Ornamental Crops Grossbeeren/Erfurt e.V., Department of Plant Quality, Theodor-Echtermeyer-Weg 1, 14979 Grossbeeren, Germany. 4Justus-Liebig University Giessen, Institute and Outpatient Clinic for Occupational and Social Medicine and Department of Anesthesiology, Intensive Care and Pain Medicine, Aulweg 129, 35392 Giessen, Germany. 5Clinic for Internal Medicine II, Molecular Biology, University of Freiburg–Medical Center, Hugstetter-Straße 55, 79106 Freiburg i.Br., Germany. 6Present address: University of Freiburg–Medical Center, Institute of Prevention and Cancer Epidemiology, Molecular Preventive Medicine, Elässerstraße 2, 79110 Freiburg, Germany. 7These authors contributed equally to this work. Correspondence and requests for materials should be addressed to S.R. (email: rohn@chemie.uni-hamburg.de)
has been done on their effect on phase II detoxification enzymes and cytotoxicity as parameters for primary and secondary/tertiary cancer prevention, respectively. A previous study compared the ability of 4-(methylsulfanyl) butyl ITC (sulforaphane) and its structural analog 5-(methylsulfanyl)pentynitrile (4-MSOB-CN, sulforaphane nitrile) to induce the activity of mammalian phase II detoxification enzymes in Fischer 344 rats. The potential of sulforaphane nitrile to induce activity of quinone reductase was less than sulforaphane, reaching the same induction as sulforaphane in a thousand-fold concentration\(^\text{12}\). In contrast, administration of the nitrile crambene (3-hydroxy-4-pentenenitrile) to Fischer 344 rats [50 mg/kg BW/day] yielded an induction of hepatic quinone reductase similar to sulforaphane\(^\text{13}\). A recent metabolism study of 3-butenenitrile (allyl-CN) showed that the cytochrome P450 isoform CYP2E1 is involved in the allyl-CN induced lethality of mice resulting from cyanide liberation\(^\text{14}\). Because of their reactivity, ITC are basically able to act also as genotoxic agents, form DNA adducts, and induce DNA mutations. Some studies even report about carcinogenic effects\(^\text{15}\). Whether these adverse effects could also be relevant for nitriles formed from GSL is not known.

The present study aimed at initially characterizing the formation of different food-borne nitriles by enzymatic hydrolysis of GSL in selected fresh \textit{Brassica} vegetables. In order to investigate structure dependent effects of nitriles \textit{in vitro}, three groups (unsaturated aliphatic nitriles, aromatic nitriles, and methylthioalkylnitriles) were selected, that belong to the most common and most important groups of glucosinolate-borne nitriles. Secondly, this study evaluated the cytotoxic and genotoxic potential of food-borne nitriles in comparison to ITC using human liver HepG2 cells as well as primary murine hepatocytes. To study whether CYP2E1 can promote hepatotoxicity, CYP2E1-overexpressing HepG2 cells were used.

**Results**

**Occurrence of glucosinolate-derived nitriles in \textit{Brassica} vegetables.** In order to investigate the relevance of plant-derived nitriles from fresh vegetables, the GSL content before and the nitrile content after autoysis of Brussels sprouts, broccoli, cauliflower, savoy cabbage, white cabbage, and red cabbage were studied. The results are presented in Table 1. A GC-MS chromatogram of nitriles from brokkoli is presented in the supplementary information. Individual GSL contents as well as the corresponding nitriles are given in \(\mu\text{mol/g fresh weight (FW)}\). Additionally, the percentage of the nitrile concentration in relation to the total content of all degradation products derived from an individual GSL are given. All \textit{B. oleracea} vegetables formed nitriles. Especially, broccoli and cauliflower were producers of significant contents of nitriles, with 57% and 48% of all detected breakdown products being nitriles. The predominant nitriles were 4-(methylsulfanyl)butynitrile (3-MSOP-CN) ranging from 0.01 (in cauliflower) to 0.43 \(\mu\text{mol/g FW}\) (in savoy cabbage), and 5-(methylsulfinyl)pentynitrile (4-MSOB-CN) ranging from 0.01 \(\mu\text{mol/g}\) (in cauliflower) to 0.32 \(\mu\text{mol/g}\) FW in broccoli, that was exceptionally rich in this nitrile. Further, broccoli and Brussels sprouts had also very high nitrile ratios (65% of the degradation products of 3-MSOB GLS were 3-MSOP-CN in broccoli) with regard to total GSL except the unsaturated aliphatic ones (Table 1). However, Brussels sprouts are also a main producer of epiphenithionitrines\(^\text{16}\).

**Cytotoxicity of food-borne nitriles in human liver (HepG2) cells.** As there is only limited information about secondary and tertiary cancer chemopreventive effects, cytotoxicity of unsaturated aliphatic, aromatic, and methylthioalkylnitriles (Fig. 1) against human liver cancer (HepG2) cells was studied. Keeping the toxicity of the related ITC in mind\(^\text{17-18}\), first investigations of the cytotoxicity of nitriles were carried out using concentrations ranging from 0.1 to 100 \(\mu\text{M}\). However, these concentrations did not impact cell viability (data not shown). Thus, a concentration range between 0.3 to 30 mM was tested. Due to a limited solubility of nitriles in purified water, no higher concentrations than 30 mM were achieved. Treatment of HepG2 cells with unsaturated aliphatic nitriles (allyl-CN, 3-but-CN and 4-pent-CN) for 72 h did not result in significant changes in cell viability. In contrast, aromatic nitriles and methylthioalkylnitriles induced a loss of viability at concentrations exceeding 10 mM (Fig. 2). However, a significant reduction of cell viability was detected only at the highest concentration tested. The cytotoxic potential of the two aromatic nitriles was similar and lead to \(\text{IC}_{50}\) values of 19.95 mM (benzyl-CN) and 18.21 mM (2-phenylethyl-CN), respectively (Table 2). Comparative investigations with benzyl-ITC resulted in a 1000-fold stronger cytotoxic potency with an \(\text{IC}_{50}\) of 15.75 \(\mu\text{M}\). Using methylthioalkylnitriles, a structure dependent inductance in cytotoxicity could be observed with increasing chain length, with the lowest \(\text{IC}_{50}\) at 8.15 mM for 6-MTH-CN. The \(\text{IC}_{50}\) for 3-MTP-CN, 4-MTB-CN, and 5-MTP-CN are given in Table 2. Figure 1 shows the structural differences of these nitriles.

**Relevance of cytochrome-P450 expression for nitrile cytotoxicity.** Recently, a study showed that the cytochrome-P450 isoform CYP2E1 is involved in the allyl-CN induced lethality of mice\(^\text{14}\). In contrast to normal human hepatocytes, the HepG2 cell line expresses only very low amounts of CYP2E1. This was verified by the present study using qPCR. Consequently and in order to assess whether the presence of CYP2E1 could further elevate the toxicity of the food-borne nitriles, the compounds were tested again using CYP2E1-overexpressing HepG2 cells. In comparison to wildtype HepG2 cells, the CYP2E1 expression in HepG2-CYP2E1 cells was 300 times higher. Results from two unsaturated aliphatic nitriles (allyl-CN, 3-but-CN), the aromatic nitrile benzyl-CN, and the methylthioalkylnitrile 4-MTB-CN are shown in Fig. 3. Only 10 mM 4-MTB-CN showed a significant difference in cytotoxicity between CYP2E1-overexpressing and vector control cells. To the best of our knowledge, there are no other proteins described in the literature that have an influence on the cytotoxicity of nitriles other than CYP2E1.

**Cytotoxicity of food-borne nitriles in normal murine hepatocytes.** To analyze the effect of nitriles on primary hepatocytes, allyl-CN, 3-butyl-CN, benzyl-CN, and 4-MTB-CN were studied in hepatocytes freshly isolated from mice. Allyl-CN, 3-but-CN and benzyl-CN significantly decreased the viability of hepatocytes in a concentration-dependent manner (Fig. 4) as analyzed by WST-1 assay after 72 h of exposure. This loss in cell
| GSL                     | nitrile (μmol/g FW) | Brussels sprouts | broccoli | cauliflower | savoy cabbage | white cabbage | red cabbage |
|------------------------|---------------------|-----------------|-----------|-------------|---------------|---------------|-------------|
| allyl-GSL              | 0.646 ± 0.046       | 0.004 ± 0.001   | 0.095 ± 0.020 | 1.292 ± 0.301 | 0.582 ± 0.193 | 0.360 ± 0.018 |
| allyl-CN               | 0.025 ± 0.008 (3%)  | n.d.            | 0.015 ± 0.011 (12%) | 0.904 ± 0.026 (9%) | 0.059 ± 0.024 (11%) | 0.019 ± 0.003 (3%) |
| 3-but-GSL              | 0.095 ± 0.010       | n.d.            | n.d.      | 0.22 ± 0.006 | 0.223 ± 0.010 | 0.224 ± 0.025 |
| 3-but-CN               | 0.008 ± 0.002 (7%)  | n.d.            | n.d.      | n.d.        | n.d.          | n.d.          |
| 2-OH-3-but-GSL         | 0.243 ± 0.041       | n.d.            | 0.002 ± 0.005 | 0.069 ± 0.027 | 0.043 ± 0.015 | 0.458 ± 0.046 |
| 2-OH-3-but-CN          | n.d.                | n.d.            | n.d.      | n.d.        | n.d.          | n.d.          |
| 3-MTP-GSL              | 0.012 ± 0.001       | n.d.            | 0.050 ± 0.019 | 0.027 ± 0.008 | 0.012 ± 0.008 | 0.010 ± 0.003 |
| 3-MTP-CN               | 0.007 ± 0.004 (100%) | 0.001 ± 0.002 (100%) | 0.10 ± 0.057 (91%) | 0.019 ± 0.008 (72%) | 0.006 ± 0.002 (30%) | 0.003 ± 0.001 (20%) |
| 4-MTB-GSL              | n.d.                | 0.013 ± 0.013   | n.d.      | n.d.        | 0.007 ± 0.010 | 0.029 ± 0.008 |
| 4-MTB-CN               | n.d.                | 0.005 ± 0.001 (47%) | 0.003 ± 0.001 (100%) | n.d.      | n.d.          | 0.005 ± 0.001 (10%) |
| 3-MSOP-GSL             | 0.707 ± 0.066       | 0.121 ± 0.013   | 0.160 ± 0.035 | 1.641 ± 0.295 | 0.609 ± 0.091 | 0.471 ± 0.061 |
| 3-MSOP-CN              | 0.188 ± 0.052 (89%) | 0.104 ± 0.154 (65%) | 0.011 ± 0.017 (41%) | 0.433 ± 0.089 (30%) | 0.177 ± 0.045 (49%) | 0.108 ± 0.012 (25%) |
| 4-MSOB-GSL             | 0.121 ± 0.019       | 0.690 ± 0.132   | n.d.      | 0.096 ± 0.020 | 0.064 ± 0.023 | 1.003 ± 0.204 |
| 4-MSOB-CN              | 0.040 ± 0.017 (93%) | 0.322 ± 0.078 (54%) | n.d.      | 0.031 ± 0.043 (49%) | 0.009 ± 0.012 (54%) | 0.256 ± 0.037 (22%) |
| benzyl-GSL             | 0.025 ± 0.006       | n.d.            | n.d.      | n.d.        | n.d.          | n.d.          |
| benzyl-CN              | 0.030 ± 0.004 (90%) | n.d.            | n.d.      | n.d.        | n.d.          | n.d.          |
| 2-Phenylethyl-GSL      | n.d.                | n.d.            | n.d.      | 0.033 ± 0.005 | n.d.          | n.d.          |
| 2-Phenylethyl-CN       | 0.007 ± 0.003 (100%) | 0.001 ± 0.001 (100%) | n.d.      | 0.006 ± 0.004 (41%) | n.d.          | n.d.          |
| 4-OH-13M-GSL           | 0.061 ± 0.013       | 0.035 ± 0.002   | 0.026 ± 0.009 | 0.012 ± 0.006 | 0.016 ± 0.007 | 0.049 ± 0.003 |
| 13M-GSL                | 0.455 ± 0.177       | 0.169 ± 0.024   | 0.252 ± 0.057 | 0.782 ± 0.365 | 0.169 ± 0.062 | 0.207 ± 0.027 |
| 1AN                    | 0.093 ± 0.030 (100%) | tr              | n.d.      | tr          | tr            | n.d.          |
| 4-Methoxy-13M-GSL      | 0.113 ± 0.025       | 0.060 ± 0.004   | 0.018 ± 0.004 | 0.049 ± 0.010 | 0.038 ± 0.016 | 0.038 ± 0.004 |
| 4-Methoxy-1AN          | 0.044 ± 0.011 (100%) | n.d.            | n.d.      | n.d.        | n.d.          | n.d.          |
| 1-Methoxy-13M-GSL      | 0.012 ± 0.009       | 0.243 ± 0.070   | 0.051 ± 0.022 | 0.012 ± 0.002 | 0.020 ± 0.011 | 0.009 ± 0.003 |
| 1-Methoxy-1AN          | 0.001 ± 0.001 (100%) | 0.007 ± 0.002 | n.d.      | n.d.        | n.d.          | n.d.          |

Table 1. Glucosinolate (GSL) content and enzymatically-formed nitriles in *Brassica* vegetables in μmol/g FW. In brackets, the amounts of formed nitriles (%CN) concerning the initial GSL content are given. Abbreviations: 3-but-GSL: 3-butenyl GSL; 2-OH-3-but-GSL: 2-hydroxy-3-butenyl GSL; 4-OH-3-but-CN: 2-hydroxy-3-butenyl nitrile; 3-MTP-GSL: 3-(methylthio)propyl GSL; 4-MTB-GSL: 4-(methylthio)butyl GSL; 3-MSOP-GSL: 3-(methylsulfinyl)propyl GSL; 4-MSOB-GSL: 4-(methylsulfinyl)butyl GSL; tr: traces; n.d.: not detected.

Risk-benefit evaluation of bioactive compounds formed in *Brassica* vegetables is not only based on its cancer preventive potency, but also on further aspects such as genotoxicity. ITC are compar-atively electrophilic compounds that have been shown to act genotoxic, at least in vitro. Representatively, the genotoxicity was investigated in terms of DNA break induction for 4-MTB-CN and benzyl-CN in comparison to benzyl-ITC using the Comet assay (Fig. 5). 4-MTB-CN and benzyl-CN induced a concentration-dependent increase in DNA strand breaks as determined by the Olive Tail Moment and percent tail DNA (Fig. 5A and B). A significant increase in DNA strand breaks was evident for 4-MTB-CN and benzyl-CN at a concentration of ≥3 μM. The DNA damage induced by 3 μM benzyl-CN was then comparable with the effect of benzyl ITC at the same concentration (Fig. 5A and B). After 24 h of recovery, the DNA damage still persisted (Fig. 5C and D). Analog to the cytotoxicity tests, next the effect of CYP2E1 expression on the genotoxic potency of nitriles was analyzed (Fig. 6). CYP2E1-overexpression caused a significant increment of DNA strand breaks at concentrations of ≥3 μM benzyl-CN as compared to vector control cells. This indicates a relevance of CYP2E1 for the genotoxic potential of this nitrile.
**Figure 1.** Chemical structure of the nitriles tested in this study.

| Structure | Name                  | Type        |
|-----------|-----------------------|-------------|
| ![Structure](image1.png) | 3-butenenitrile | unsaturated aliphatic |
| ![Structure](image2.png) | 4-pentenenitrile | unsaturated aliphatic |
| ![Structure](image3.png) | 5-hexenenitrile | unsaturated aliphatic |
| ![Structure](image4.png) | phenylacetonitrile | aromatic |
| ![Structure](image5.png) | 3-phenylpropanenitrile | aromatic |
| ![Structure](image6.png) | 4-(methylthio)butynitrile | methylthioalkyl |
| ![Structure](image7.png) | 5-(methylthio)pentynitrile | methylthioalkyl |
| ![Structure](image8.png) | 6-(methylthio)hexynitrile | methylthioalkyl |
| ![Structure](image9.png) | 7-(methylthio)heptylnitrile | methylthioalkyl |

**Figure 2.** Analysis of cell viability using the WST-1 assay after a 72 h treatment with unsaturated (A) aromatic nitriles (B), methylthioalkynitriles (C) and benzyl-ITC (D). Control: solvent control, 0–30% double distilled sterile water (A–C), 0.1% DMSO (D). Bars are mean ± SEM, n = 3. *P < 0.05; asterisks indicate a significant difference between the sample and the solvent control.
Discussion

Many Brassica vegetables contain alkenyl-GSL that can be hydrolyzed to ITCs, nitriles, and epithionitriles. However, studies on the occurrence of food-borne nitriles and the knowledge about their biological effects are still not comprehensively. In the present study, the predominant nitriles in vegetables were 3-MSOP-CN and 4-MSOB-CN, the latter ranging from 0 μmol/g (in cauliflower) to 0.32 μmol/g FW in broccoli that was exceptionally rich in this nitrile. This is in line with data reported in the literature. In the study of Matusheski et al. (2004), 4-MSOB-CN concentrations ranged from 0.35 to 1.50 μmol/g fresh weight (in broccoli). Moreover, the preferential formation of 3-MSOB-CN and 4-MSOB-CN compared to the corresponding ITC has been already observed by Daxenbichler et al. 21 for freshly autolyzed cabbage.

In the actual study, Brussels sprouts had also very high nitrile ratios with regard to total GSL, except the unsaturated aliphatic ones. The latter released, probably due to a high ESP activity, the corresponding epithionitriles and therefore Brussels sprouts were in total a producer of epithionitriles which is in agreement to others 21. However, the highest total nitrile concentration with 0.58 μmol/g FW was found in savoy cabbage, although it was, likewise to red cabbage, also a well-known producer of ITC. This was due to a high content of 3-MSOP-GSL and its hydrolysis products comprised to 37% of the nitrile. Therefore, consumption of 200 g savoy cabbage would mean an ingestion of 86 μmol of this nitrile. Moreover, it has been shown that the aliphatic nitrile allyl-CN

| nitriles       | IC50 [mM] |
|----------------|-----------|
| benzyl-CN      | 19.95     |
| 2-phenylethyl-CN| 18.21     |
| 3-MTP-CN       | 19.49     |
| 4-MTB-CN       | 17.89     |
| 5-MTP-CN       | 12.19     |
| 6-MTH-CN       | 8.15      |

Table 2. Calculated median growth inhibitory concentration of aromatic and methylthioalkylnitriles after 72 h treatment as determined in the WST-1 assay.

Figure 3. Analysis of cell viability of CYP2E1 transfected HepG2 cells and vector cells after treatment with nitriles. Cells were treated with allyl-CN (A), 3-but-CN (B), benzyl-CN (C) and 4-MTB-CN (D) for 72 hours and viability was analyzed using WST-1 assay. Bars are mean ± SEM, n = 3. *P ≤ 0.05, asterisks indicate a significant difference between the sample and the solvent control: medium. CYP2E1: HepG2-CYP2E1, vector: HepG2-vector.
resulting from allyl-GSL occurred, in all of the B. oleracea vegetables investigated, except in broccoli. Consistently, in all analyzed B. oleracea vegetables also 3-MTP-CN was formed. The occurrence of this nitrile has been already reported for cabbage and cauliflower21,22. However, during the last decades, ITC were made responsible for any physiological effects of Brassica consumption, e.g., also for cancer prevention. Therefore, it is mandatorily necessary to also clarify possible effects of food-borne nitriles.

As shown by Matusheski and Jeffery12, the ability of the sulforaphane nitrile to induce quinone reductase was about 1000-fold less than that of sulforaphane itself. Sulforaphane induced quinone reductase at a concentration of 2.5 μM, reaching a maximum induction of 3.1-fold, whereas the corresponding nitrile reached a similar induction only at a concentration of 2 mM12. In a recently published study, strong interference of benzyl-ITC at ≥3 μM with the cellular pro-inflammatory response including prostaglandin PGE2 release was shown in activated human T lymphocytes. In contrast, benzyl-CN did not influence PGE2 release even at a concentration of 1 mM23. These results confirm the observation that nitriles affect their target less than their corresponding ITC.
when applied at similar concentrations. In the present study, the cytotoxic potential of two aromatic nitriles (benzyl-CN and 2-phenylethyl-CN) was comparable. Investigations with benzyl-ITC resulted in a 1000-fold stronger cytotoxic potency compared to the nitriles (Fig. 2D). Lui et al. (2003) determined the IC_{50} of benzyl-ITC using the MTT-assay in the head and neck squamous carcinoma cell line UM-2BB with a value of 17 μM. These data imply that in comparison to ITC, food-borne nitriles are far less toxic to human cancer cells in vitro. Indications are given that a minor structural difference such as the loss of a methyl group, could lead to different structurally-dependent effects being involved in cytotoxicity mechanism. Ahmed & Farooqui (2011) reported for rats that the toxicity of aliphatic nitriles not only diversifies with cyanide liberation but is also highly dependent on their chemical structure. In that study, saturated nitriles at amounts of 0.7 mmol produced symptoms of poisoning similar to the symptoms produced by inorganic cyanide (0.15 mmol). In contrast, unsaturated nitriles with amounts of 1.7 mmol showed only moderate symptoms. In the present study, methylthioalkylnitriles with increasing chain length lead to a structure-dependent increase in cytotoxicity, with 6-MTH-CN being the most toxic substance (Table 2). It has to be taken into account that concentrations inducing cell toxicity were still very high. In a previous study, the IC_{50} of 4-MTB-ITC was determined using the neutral red retention assay with 23.18 μM in HepG2 cells. This ITC, again, is a 1000-fold more potent in inducing cell death as compared to the cytotoxicity observed with the nitrile 4-MTB-CN in the present study. In accordance to others, data of the present study also imply that no relevant cytotoxicity occurs when an aliphatic moiety in the nitrile is present. However, presence of sulfur or an aromatic ring in the chemical structure of the chain led to an increased cytotoxicity by 76%. Of those, cell viability was reduced by 12%. Thus, tested concentrations in this study are comparably high and not expected to occur in vivo after consumption of processed or non-processed Brassica foods.

Nitriles seem only to be activated by CYP2E1 and other CYPs. Thus, it was suggested that the toxic effects result from cyanide liberation following the activation of certain enzymes. Although cytotoxicity of nitriles was not affected by CYP2E1-overexpression in HepG2 cells in the present study, this was the case for genotoxicity. Here, a genotoxic effect was already significant at a concentration of ≥3 μM. Moreover, CYP2E1 overexpression increased genotoxicity. Whereas ITC are well known genotoxic agents in vitro, this was not known for nitriles. For example Lamy et al. (2009) investigated the genotoxic potential of 3-MTP-ITC, 4-MTB-ITC, and 5-(methylthio)pentyl ITC (5-MTP-ITC) in HepG2 cells using the Comet assay as well as the micronucleus test. That study concluded that the ITC induced DNA damage and mutagenicity at concentrations exceeding 1 μM. Based on the present data nitriles seem to be as potent as ITC to induce DNA damage and this damage seems to be persistent. The intensity of DNA strand breaks of 3 μM benzyl-CN was comparable with the effect of benzyl-ITC in the same concentration. In vitro experiments with benzyl-ITC showed characteristic concentration-dependent genotoxic effects under different conditions (dependent on the cell line or endpoint). Comparatively weaker effects were observed in in vivo experiments. The genotoxicity of benzyl-ITC was reduced by bovine serum albumin and human body fluids, which suggests that benzyl-ITC is inactivated under in vivo conditions. If nitriles are also detoxified in vivo, needs to be clarified in future studies.

Under the conditions applied in the present study, B. oleracea vegetables released high amounts of nitriles upon hydrolysis of GSL. Especially broccoli and cauliflower were intense producers of nitriles. The highest total nitrile concentration was detected in savoy cabbage, though comprising mainly of 3-MSOP-CN. With regard to cell culture experiments, tested nitriles were cytotoxic towards cancer cells only when being applied in 1000-fold concentrations.
higher concentrations, compared to the ITC. Thus, an impact of nitriles for a secondary and tertiary cancer preventive potential by *Brassica* vegetables seems negligible. Moreover, with regard to common habits of food intake, an indication for a preventive potential of GSL-derived nitriles can still not be given with the present data. However, the tested nitriles showed genotoxic potential in HepG2 cells, comparable in effect to their corresponding ITC. Thus, the preference of nitrile instead of ITC formation under certain food processing conditions should be taken into account. Moreover, the weak cytotoxic activity of nitriles while being quite potent genotoxic agents should be considered carefully, as well. This genotoxic activity of nitriles, as shown in this study, can be considered as a risk factor in cancer prevention. Thus, this could be one relevant factor being involved in the contradictory observations made with cancer prevention studies using *Brassica* vegetables. As only three structure-types of nitriles were tested in the present study, further research with nitriles differing in their chemical structure is necessary. Based on the in vitro data presented, there are no indications that nitriles affect health without a preceding activation by CYP enzymes. However, with respect to genotoxicity data, it cannot be ruled out that nitriles pose a risk under conditions relevant for food consumption.

**Materials and Methods**

**Chemicals.** Benzonitrile (>99.9%), 3-phenylpropanenitrile (2-phenylethyl-CN, ≥99%), 3-butenenitrile (allyl-CN, ≥98%), 4-pentenenitrile (3-but-CN, ≥97%), 2-propenyl GSL hydrate (sinigrin, alllyl-GSL, ≥99%), 2-propenyl ITC (allyl-ITC, ≥99%), 3-(methylthio)propyl ITC (3-MTP-ITC, ≥98%), 4-bromobutrynitrile (≥97%), 6-bromohexanenitrile (≥95%), 7-bromohexanenitrile (≥90%), triton X-100 (laboratory grade), and trypan blue powder were purchased from Sigma-Aldrich GmbH (Steinheim, Germany); benzyl ITC (benzyl-ITC, ≥97%) and phenylacetoniitrile (benzyl-CN, ≥98%) from Lancaster Synthesis Ltd. (Morecambe, UK); 3-indoleacetonitrile (IAN, ≥98%) from Acros Organics (Fischer Scientific GmbH, Schwerte, Germany); 5-hexenonitrile (4-pent-CN, ≥95%), 3-butenyl ITC (3-but-ITC, ≥95%), 4-butenyl ITC (4-pent-ITC, ≥95%) and 5-bromovaleronitrile (≥95%) were purchased from TCI Deutschland GmbH (Eschborn, Germany); 3-(methylsulfinyl)propyl ITC (3-MSOP-ITC, ≥97%) and 4-(methylthio)butyl ITC (4-MTB-ITC, ≥98%) were purchased from Santa Cruz Biotechnology (Heidelberg, Germany); 4-(methylsulfonyl)butyl ITC (4-MSOB-ITC, ≥98%) was purchased from Enzo Life Sciences GmbH (Lörrach, Germany); 1-cyano-2,3-epithiopropane (CETP, ≥95%) was purchased from Taros Chemicals GmbH Co. KG (Dortmund, Germany); methanol (>99.9) and Na₂SO₄ were purchased from VWR International GmbH (Darmstadt, Germany); methylene chloride (GC Ultra Grade) was purchased from Carl Roth GmbH (Karlsruhe, Germany); diethyl ether (>99.9%) was purchased from Teijin Chemicals (Pasching, Austria) and Gibco®, Life Technologies GmbH (Darmstadt, Germany). Penicillin-Streptomycin solution was purchased from Life Technologies GmbH (Darmstadt, Germany). Acetonitrile (Ultra Gradient HPLC grade) was purchased from J.T. Baker (Deventer, Netherlands). RNasey mini isolation kit and RNase-free DNase kit were purchased from QiAGEN GmbH (Hilden, Germany). Absolut™ QPCR SYBR Green Capillary Mixes was purchased from Fisher Scientific (Pittsburgh, PA, USA).

**Chemical synthesis of methylthioalkynitriles.** Due to the poor commercial availability of methylthioalkynitriles, the four tested compounds have been synthesized according to Moon et al.²⁶: 4-(methylthio)butynitrile (3-MTP-CN). Sodium thiomethoxide (0.23 g, 3.3 mmol; 95%, Sigma-Aldrich GmbH, Steinheim, Germany) was dissolved in 5 mL methanol and gradually added to a solution of 4-bromobutrynitrile (0.5 g, 3.4 mmol; 97%) in 5 mL methanol. The mixture reacted 7 h under reflux conditions. Subsequent to evaporating the methanol in vacuo, the resulting product was dissolved in 13 mL water and extracted with diethyl ether (3 × 25 mL). The combined organic extracts were dried over sodium sulfate and the solvent was removed. 5-(methylthio)pentynitrile (4-MTB-CN), 6-(methylthio)hexynitrile (5-MTP-CN) and 7-(methylthio)heptynitrile (6-MTH-CN) were synthesized as described in the section Chemical synthesis; trypsin, trypsin/EDTA, low glucose Dulbecco’s modified eagle medium, fetal calf serum and L-glutamine (200 mM) were purchased from PAA Laboratories GmbH (Pasching, Austria) and Gibco®, Life Technologies GmbH (Darmstadt, Germany). Penicillin-Streptomycin solution was purchased from Life Technologies GmbH (Darmstadt, Germany). Acetonitrile (Ultra Gradient HPLC grade) was purchased from J.T. Baker (Deventer, Netherlands). RNasey mini isolation kit and RNase-free DNase kit were purchased from QiAGEN GmbH (Hilden, Germany). Absolut™ QPCR SYBR Green Capillary Mixes was purchased from Fisher Scientific (Pittsburgh, PA, USA).

**Plant Material and Sample Preparation.** For determining the occurrence of nitriles in vegetables, selected *Brassica* vegetables were grown at the experimental sites of the Leibniz Institute of Vegetable and Ornamental Crops Großbeeren/Erfurt e.V., Großbeeren, Germany and harvested during autumn 2013. Fertilization, irrigation, and plant protection corresponded to standard cultivation procedures of *Brassica* vegetables. As economically important horticultural *Brassica* species, white cabbage (*Brassica oleracea* var. capitata f. alba, cv. Reaktion), red cabbage (*Brassica oleracea* var. capitata f. rubra, cv. Pesaró), savoy cabbage (*Brassica oleracea* var. sabauda, cv. Traviata), cauliflower (*Brassica oleracea* var. botrytis, cv. Chambord), broccoli (*Brassica oleracea* var. italica, cv. Iron Man), and Brussels sprouts (*Brassica oleracea* var. gemmifera, cv. Crispus) were chosen. For determining the amount of the different compounds, fully developed heads (cabbages, broccoli, and cauliflower) or fully developed leaf buds (Brussels sprouts) were harvested. For each *Brassica* vegetable, five biological replicates were taken using a representative aliquot for the determination of GSL and nitriles. Vegetables were free of injuries or pests and analyzed without delay after harvest. *Brassica* head forming and inflorescence vegetables were cut into 5 × 5 cm pieces (cabbages, broccoli, cauliflower), the leaf buds of Brussels sprouts were cut in half and mixed. Half of each sample was immediately frozen at −50 °C and lyophilized for GSL determination. From the other half, an aliquot was crushed with a homogenizer (Edmund Bühler H04, Tübingen, Germany) after
addition of an aliquot of water (one part of the plant/one part of water). From the homogenized sample an aliquot (1 g containing 500 mg of water) was immediately subjected to the analysis of nitriles.

**Cell culture.** HepG2 cells were purchased from the German Collection of Microorganisms and Cell Cultures (DSMZ, Braunschweig, Germany). HepG2 cells which constitutively express human CYP2E1 (E47 cells, HepG2-CYP2E1) and control HepG2 cells transfected with empty vector (C34 cells, HepG2- vector), originally established by Wu and Cederbaum 27, were kindly provided by C. Hellerbrand (University Medical Center Regensburg, Germany). The cells were cultured in low glucose Dulbecco's modified eagle medium supplemented with 15% fetal calf serum and 1% Pen/Strep in an atmosphere with 5% CO2 at 37°C and 95% air humidity.

**Determination of the drug effect in primary murine hepatocytes.** Murine hepatocytes were obtained from 8–16 week-old, male C57BL6 mice (Charles River Laboratory, Sulzfeld, Germany) as described elsewhere 24. The institutional and national guidelines for the care and use of animals were followed, and all experimental procedures were approved by the institutional Animal Care and Use Committee at the University of Freiburg according to § 8.1 of the Animal Welfare Act. 2–2.2 × 105 cells were plated in 24 well plates. The cells were left undisturbed for 4 h in a 5% CO2 atmosphere at 37°C and then the medium was changed to serum free Williams' Medium E supplemented with 100 nM dexamethasone. Within 24 h the cells were exposed to the test compounds and controls for 72 h.

**Determination of glucosinolates from Brassica vegetables.** GSL concentration in vegetable samples was determined as desulfo-GSL using the DIN EN ISO 9167–1 based method as previously described in ref. 28. Briefly 15 mg of lyophilized plant material were extracted methanolically, GLS were desulfated and analyzed by UHPLC-DAD using a Poroshell 120 EC-C18 column, 100 mm × 2.1, particle size 2.7 μm (Agilent Technologies and a gradient comprising of water and acetonitrile as reported previously 28. Desulfo-GSL were identified by comparing retention times and UV absorption spectra with those of authentic standards. Quantification was done by using an external calibration curve with allyl-GSL and a wavelength of 229 nm.

**Determination of nitriles in Brassica vegetables.** For the determination of nitriles, the method reported previously by Witzel et al. (2013) was adapted. An aliquot of vegetable sample (1 g consisting of 500 mg plant material and 500 mg of water) was weighted into a centrifugal tube, benzonitrile as internal standard (IST) was added, analytes were extracted with methylene chloride, and analyzed by GC-MS as described previously 29 using a SGE BPSMS column 30 m × 0.25 mm × 0.25 μm (VWR International GmbH, Darmstadt, Germany). Analytes were identified by comparing mass spectra and retention times with those of authentic standards. Quantification was done by using an external calibration curve with allyl-GSL and a wavelength of 229 nm.

**Determination of cytotoxicity.** Cytotoxicity was determined with the WST-1 assay (Roche Diagnostics Deutschland GmbH, Mannheim, Germany). The assay is based on the ability of viable cells to cleave the tetrazolium salt WST-1 to formazan by their mitochondrial succinate-tetrazolium-reductase system. Cells were cultured at an initial density of 5000 cells/well in 96-well plates for 48 h. Then, cells were treated with the nitriles (0.3 to 30 mM, solved in distilled water) for 72 h. As positive control, 0.01% Triton X-100 was used. Following compound exposure and incubation, the WST 1 assay was performed according to the manufacturer's instructions.

**Single cell gel electrophoresis assay (Comet Assay).** The single cell gel electrophoresis assay also known as Comet assay was carried out as described with slight modifications 30. The percent tail DNA and the Olive Tail Moment (OTM) were calculated as indicators of DNA damage. For each sample, 102 systematically screened cells were evaluated. Briefly, HepG2 cells were seeded onto well plates. After 48 h of growth, the cells were exposed to the nitriles or ITC for 24 h. As positive control 50 μM B(α)P and as solvent control distilled water or 0.1% DMSO was used. To analyze the repair capacity of HepG2 cells, the test substances were removed after 24 h and incubated for another 24 h before the assay was terminated.

**Real-time polymerase chain reaction.** Total RNA was isolated with the RNeasy mini Isolation kit from QIAGEN GmbH followed by a purificaion step using the RNase-free DNase kit from QIAGEN GmbH according to manufacturer’s instructions. 5 μg RNA of each cell line was dried by a vacuum concentrator (Eppendorf AG, Hamburg, Germany). Isolated RNA was re-suspended in 10 μL of RNase-free water. Each sample was treated twice with 2 μL RNase-free DNase 1 unit/μL (QIAGEN GmbH) for 10 min at 37°C to eliminate remaining DNA. The RNA prepared was reverse-transcribed as described elsewhere 31.
For quantitative comparison of CYP2E1 mRNA levels real-time polymerase chain reaction (qPCR) was performed using SYBR-green fluorescence in a LightCycler® System (Roche Diagnostic GmbH, Mannheim, Germany). After optimization of the PCR conditions, amplification efficiency was tested in standard curves using serial cDNA dilutions. The correlation coefficient had to be above 0.9 and the slope around –3.5. Amplification specificity was checked using melting curves. Gene expression was related to the mean expression of the four housekeeping genes (HSK) beta-2-microglobulin (B2M), beta-actin (β-Actin), hypoxanthine-guanine phosphoribosyltransferase (HPRT) and 5-monooxygenase activation protein, zeta (YWAHZ). Calculations of expression was performed with the 2−ΔΔCT method according to Pfaffl et al.32 The sequence of the specific primers were CYP2E1f (5′-CCT ACA TGG ATG CTG TGG TG-3′), CYP2E1rev (5′-TGG GGA TGT ATC CTC TG-3′), B2Mf (5′-ACTGAATTCGCCCTGGA-3′), B2Mrev (5′-ACTCCATGATGCGTTACA-3′), β-Actinf (5′-CTG GAA CGG TGA AGG TA-CA-3′), β-Actinrev (5′-AAG CTT CCT GCT GTA ACG CA-3′) HPRTf (5′-ATG CTG AGG ATT TGG AGG GGA-3′), HPRTrev (5′-GCA CAC AGA GGA GTG CCA AA-3′) and YWAHZf (5′-ACT TTT GGT ACA TTT GGG CTT CAA-3′) YWAHZrev (5′-CGG CCA GGA GAA ACC AGT AT-3′). PCR reactions were carried out in a final volume of 20 μL using 1x Absolut™ QPCR SYBR Green Capillary Mixes, 300 nM of HSK primers or 600 nM of CYP2E1 primers and 2 μL cDNA. The PCR conditions for CYP2E1 were as follows: Initial denaturation 15 min at 95 °C, touch down PCR 2 cycles of 95 °C for 10 s, 65 °C for 10 s, 72 °C for 25 s; 2 cycles of 95 °C for 10 s, 62 °C for 10 s, 72 °C for 25 s; 45 cycles of 95 °C for 10 s, 61 °C (B2M), 61 °C (HPRT), 56 °C (β-Actin), 58 °C (YWAHZ) for 15 s, 72 °C for 15 s. All measurements were made without information about the origin of the samples and were performed in duplicate.

Statistical Analysis. The median growth inhibitory concentration (IC50) for compound cytotoxicity was calculated using Graphpad Prism 5.0 software (La Jolla, CA). After logarithmic transformation, IC50 values were calculated with equation 1:

\[ y = 100/1 - 100/\left(\log IC_{50} - x\right) \times \text{Hillslope} \]  

(1)

All experiments were performed using four replicates and were repeated at least two times, independently. Significance of difference between the treatment groups in the Comet assays was analyzed using the ordinary one-way ANOVA33 and WST-1 data was analyzed by using the one way ANOVA followed by Bonferroni correction. The comparison of the two groups was analyzed using an unpaired t-test. The differences for both tests with p ≤ 0.05 (*) were considered statistically significant.

References
1. Kim, J. K., Shin, D.-H., Park, H. G. & Shin, E.-C. Cruciferous vegetables, glutathione S-transferases, and implications of their interaction to colorectal cancer risk: A review. Journal of the Korean Society for Applied Biological Chemistry 57, 511–517, doi: 10.1007/s11765-014-0104-3 (2014).
2. Hanschen, F. S., Lamy, E., Schreiner, M. & Rohn, S. Reactivity and stability of glucosinolates and their breakdown products in foods. Angewandte Chemie International Edition 53, 11430–11450, doi: 10.1002/anie.201402639 (2014).
3. Lamy, E. & Mersch-Sundermann, V. MTRBITAN generates cell cycle arrest and apoptosis induction in human HepG2 cells despite its rapid degradation kinetics in the in vitro Model. Environ. Mol. Mutagen. 50, 190–200, doi: 10.1002/em.20448 (2009).
4. Wiesner, M., Schreiner, M. & Glatt, H. High mutagenic activity of juice from pak choi (Brassica rapa ssp. chinensis) sprouts due to its content of 1-methoxy-3-indolylmethyl glucosinolate, and its enhancement by elicitation with methyl jasmonate. Food Chem. Toxicol. 47, 10–16, http://dx.doi.org/10.1016/j.fct.2008.06.001 (2009).
5. Oerlemans, K., Barrett, D. M., Suades, C. B., Verkerk, R. & Dekker, M. Thermal degradation of glucosinolates in red cabbage. Food Chemistry 1685–1707, doi: 10.1002/mnfr.201300684 (2014).
6. Handschen, F. S., Rohn, S., Mewis, I., Schreiner, M. & Kroh, L. W. Influence of the chemical structure on the thermal degradation of the glucosinolates in broccoli sprouts. Food Chemistry 130, 1–8, doi: 10.1016/j.foodchem.2011.05.109 (2012).
7. Gupta, P., Kim, B., Kim, S.-H. & Srivastava, S. K. Molecular targets of isothiocyanates in cancer: Recent advances. Mol. Nutr. Food Res. 58, 1685–1707, doi: 10.1002/mnfr.201300684 (2014).
8. Barrera, L. N., Cassidy, A., Johnson, I. T., Bao, Y. P. & Belshaw, N. J. Epigenetic and antioxidant effects of dietary isothiocyanates and selenium: potential implications for cancer chemoprevention. Proc. Nutr. Soc. 71, 235–245, doi: 10.1017/s002966511200016x (2012).
9. Williams, D. J., Critchley, C., Pun, S., Chalhia, M. & O’Hare, T. J. Differing mechanisms of simple nitrile formation on glucosinolate degradation in Lepidium sativum and Nasturtium officinale seeds. Phytochemistry 70, 1401–1409, http://dx.doi.org/10.1016/j.phytochemistry.2009.07.035 (2009).
10. Wittstock, U. & Burrow, M. Glucosinolate Breakdown in Arabidopsis: Mechanism, Regulation and Biological Significance. The Arabidopsis Book 6, doi: 10.1199/tab.0134 (2010).
11. Hanschen, F. S. et al. Thermally induced degradation of sulfur-containing aliphatic glucosinolates in Broccoli Sprouts (Brassica oleracea var. italica) and model systems. J Agr Food Chem 60, 2231–2241, doi: 10.1021/jf0204830p (2012).
12. Matushkesi, N. V. & Jeffery, E. H. Comparison of the bioactivity of two glucoraphanin hydrolysis products found in broccoli, sulforaphane and sulforaphane nitrile. J Agr Food Chem 49, 5734–5749, doi: 10.1021/jf010809a (2001).
13. Keck, A. S., Staack, R. & Jeffery, E. H. The cruciferous nitrite catabolism has bioactivity similar to sulforaphane when administered to Fischer 344 rats but is far less potent in cell culture. Nutr Cancer 42, 233–240, doi: 10.1207/s15327914nc42_13 (2002).
14. Boadas-Vaello, P. et al. Alkylthiolethyl metabolism by CYP2E1 and other CYPs leads to distinct lethal and vestibulotoxic effects in the mouse. Toxicol. Sci. 107, 461–472, doi: 10.1093/toxsci/kfn233 (2009).
15. Fimognari, C., Turrini, E., Ferruzzi, L., Lenzi, M. & Hrelia, P. Natural isothiocyanates: Genotoxic potential versus chemoprevention. Mutat. Res.-Rev. Mutat. Res. 750, 107–151, doi: 10.1016/j.mrrev.2011.12.001 (2012).
16. Hanschen, F. S. et al. Brassica epithionitrile 1-cyano-2,3-epithiopropane triggers cell death in human liver cancer cells in vitro. Mol. Nutr. Food Res. 59, 2178–2189, doi: 10.1002/mnr.21090296 (2015).
17. Cohen, J. H., Kristal, A. R. & Stanford, J. L. Fruit and vegetable intakes and prostate cancer risk. Journal of the National Cancer Institute 92, 61–68, doi: 10.1093/jnci/92.1.61 (2000).
18. Lamy, E. et al. Preclinical evaluation of 4-methylthiobutyl isothiocyanate on liver cancer and cancer stem cells with different p53 status. Plos One 8, doi: 10.1371/journal.pone.0070846 (2013).
Cytotoxic and genotoxic potential of food-borne nitriles in a liver: Kupke, F. et al.

How to cite this article

Kupke, F. et al. Cytotoxic and genotoxic potential of food-borne nitriles in a liver. Scientific Reports 6:37631; doi: 10.1038/srep37631 (2016).

Additional Information

Supplementary information accompanies this paper at http://www.nature.com/srep

Competing financial interests: The authors declare no competing financial interests.

How to cite this article: Kupke, F. et al. Cytotoxic and genotoxic potential of food-borne nitriles in a liver in vitro model. Sci. Rep. 6, 37631; doi: 10.1038/srep37631 (2016).

Publisher's note: Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

This work is licensed under a Creative Commons Attribution 4.0 International License. The images or other third party material in this article are included in the article’s Creative Commons license, unless indicated otherwise in the credit line; if the material is not included under the Creative Commons license, users will need to obtain permission from the license holder to reproduce the material. To view a copy of this license, visit http://creativecommons.org/licenses/by/4.0/.

© The Author(s) 2016