Comparative Evaluation of Sulfur Compounds Contents and Antiobesity Properties of *Allium hookeri* Prepared by Different Drying Methods

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Despite the nutritional and medicinal values of *Allium hookeri*, its unique flavor (onion or garlic taste and smell) coming from sulfur containing compounds limits its usage as functional food. For comparative study, *A. hookeri* roots were prepared under two different drying conditions, namely, low-temperature drying that minimizes the volatilization of sulfur components and hot-air drying that minimizes the garlic odor and spicy taste of *A. hookeri*. In GC/MS olfactory system, the odorous chemicals and organosulfur compounds such as diallyltrisulfide, dimethyltrisulfide, and dipropyltrisulfide were significantly decreased in hot-air drying compared to low-temperature drying. The spiciness and saltiness taste were noticeably reduced, while sourness, sweetness, and umami taste were significantly increased in hot-air dried *A. hookeri* according to electronic tongue. Although the content of volatile sulfur components was present at lower level, the administration of hot-air dried *A. hookeri* extract (100 mg/kg p.o.) apparently prevented the body weight gain and improved insulin resistance in C57BL/6j obese mice receiving high fat diet. Results suggested that the hot-air dried *A. hookeri* possessing better taste and odor might be available as functional crop and bioactive diet supplement for the prevention and/or treatment of obesity.

1. Introduction

The species of *Allium* including *Allium cepa* (onions), *Allium sativum* (garlic), and *Allium schoenoprasum* (chives) have been consumed as medicinal food for a long time [1]. Phytochemically, *Allium* plants contain sulfur compounds as main bioactive constituents [2]. The unique taste and odor of *Allium* species comes from the enzymatic hydrolysis of several precursors of sulfur-containing flavor compounds [3]. Allicin (diallyl thiosulfinate) is a main allyl sulfur component of garlic (*Allium sativum* Linn.) that constitutes over 60% of total garlic thiosulfinates. Thiosulfinate including allicin is a group of very unstable compounds. Once it is generated, it readily gives rise to further rearrangement leading to the production of a variety of derivatives. It has been reported that the amount of organosulfur compounds in the genus *Allium* can be changed depending on the drying conditions such as temperature, time, or an apparatus used [4, 5]. *Allium hookeri* Thwaites (Liliaceae) is a terrestrial perennial herb and widely grown as medicinal food in Southeast Asia countries [6]. The root of *A. Hookeri*, also called “juumyit” or “napakpi,” has been traditionally used to treat
inflammatory diseases in India and Myanmar [7]. The A. hookeri is known to have biological properties such as antioxidant, anticancer, and anti-inflammatory effects [7–9]. The therapeutic benefits of A. hookeri are attributed to the presence of diverse phenols, phytoesterols, and organosulfur compounds [10]. A. hookeri contains a strong flavor which comes from the high content of diallyl sulfide, a flavor component of garlic (A. sativum). It has been reported that diallyl sulfide is contained in the A. hookeri with a higher level of abundance than in the A. sativum [9].

Obesity arises from dysregulation of energy balance due to excessive energy intake and insufficient energy expenditure [11]. Epidemic increase in overweight and obesity is of medical concern because it is an important risk factor for several chronic diseases, particularly dyslipidemia and type 2 diabetes [12, 13]. Excessive body weight gain has been intimately associated with insulin resistance in individuals with type 2 diabetes [14]. Besides, obesity is a strong causal factor for sleep-disordered breathing, which contributes to the increased cardiovascular mortality [15]. Naturally occurring products have attracted researchers’ attentions as sources of new drugs and drug leads for the treatment of obesity [16–18]. The species of Allium have been used as a folk medicine for the treatment of metabolic diseases and Allium-derived extracts have recently been of interest for their antiobesity effects [19, 20]. However there has not been any attempt to reveal the beneficial effects of A. hookeri in prevention and/or treatment of obesity.

Despite important therapeutic and culinary value, people avoid the A. hookeri because of their unique odor and spicy taste induced by volatile sulfur compounds [21]. Therefore, we attempted to develop the optimized drying methods of A. hookeri that minimize the unpleasant odor and taste without reducing its biological activity. Two dried A. hookeri have been prepared under the different drying operation, namely, low-temperature drying and hot-air drying. The contents of organosulfur, odorous components, were compared, and antioxidant, anticancer, and anti-inflammatory effects [7–9]. Therefore, we attempted to develop the optimized drying methods of A. hookeri that minimize the unpleasant odor and taste without reducing its biological activity. Two dried A. hookeri have been prepared under the different drying operation, namely, low-temperature drying and hot-air drying. The contents of organosulfur, odorous components, were compared, and sensory test has been performed. Also, in vivo test was undertaken to ascertain the antiobesity effect of two types of A. hookeri in high-fat diet-induced C57BL/6J obese mouse model.

2. Materials and Methods

2.1. Plant Material and Reagents. A. hookeri was cultivated and harvested in Hadong (Korea). The roots of A. hookeri were washed three times with five volumes of distilled water. A voucher specimen was deposited in Laboratory of Pharmacognosy in Gyeongnam National University of Science and Technology.

2.2. Drying Procedures and Sample Preparation. A. hookeri roots were dried for 96 h in an oven at 40°C to minimize the volatilization of sulfur compounds contained. The darkened A. hookeri roots were autoclaved for 10 min at 105°C, then steamed for 30 min at 80°C, and further steamed for 24 h at 70°C followed by drying for 72 h at 35°C. A Likens and Nickerson-type simultaneous steam distillation and extraction apparatus (SDE) was used for the extraction of volatile compounds. Ground samples (100 g) were mixed with 1L of distilled water followed by the addition of internal standard (1mL of pentadecane, 1mg/mL, Sigma-Aldrich Co.). Atmospheric steam distillation was performed to collect volatile oils from the sample in a 100 mL mixture of pentane and diethyl ether (1:1, v/v) at 110°C for 3 h. Anhydrous sodium sulfate (10 g) was added to the extract, which was then maintained at 4°C overnight. Samples were then filtered and reduced to a volume of 1 mL using a nitrogen evaporator.

2.3. Gas Chromatography-Mass Spectrometry (GC-MS) Analysis. The essential oil was analyzed using an Agilent 7890A and 5975C (Agilent Technologies) and HP-5MS capillary column (30 m × 0.25 mm × 0.25 μm film thickness). Injector and detector temperatures were set at 220°C and 290°C, respectively. Column temperature was initially kept at 40°C for 5 min and then gradually increased to 200°C at a rate of 5°C/min. Helium was used as carrier gas at a flow rate of 1 mL/min. Samples of 1 μL were injected manually in the splitless mode. Tentative identification of the compounds was based on the comparison of their mass spectra with those of NIST 98 and Wiley 275 library data of the GC/MS system. Quantitative data were obtained electronically from FID area percent data without the use of correction factors.

2.4. Gas Chromatography-Olfactometry. In parallel with GC-MS, samples were analyzed by GC/MS-olfactory detection port with heated mixing chamber (ODP 3, Gerstel, Inc., Linthicum, MD, USA) using the concept of detection frequency. A sniffing procedure panel was formed of 3 judges who were chosen from 10 assessors well trained in sensory evaluation. Results of GC/MS-O analyses were expressed as average values of odor intensity in a scale from 1 to 7 with increments of 1, obtained from 3 independent measurements.

2.5. Sensory Evaluation Test. The Astree II e-Tongue system, developed by Alpha M.O.S. (Toulouse, France), was used for taste evaluation of the test example solutions. Astree II is a fully automated taste analyzer equipped with seven sensors, ZZ, AB, GA, BB, CA, DA, and JE, based on the ChemFET technology (Chemical Modified Field Effect Transistor) for liquid samples analysis (Alpha M.O.S., 2004). In the presence of dissolved compounds, a potentiometric difference is measured between each of the seven sensors and the Ag/AgCl reference electrode. Each sensor has a specific organic membrane, which interacts with ionic, neutral, and chemical compounds present in the sample solution in a specific manner. Any interaction at the membrane interface is detected by the sensor and converted into an electronic signal. The raw data is expressed as voltage versus time. For these experiments, only the last 20 s of the 120 s data was used in the analysis. Samples were replicated five times and the average value of the last four measurements is used in the data analysis. The sensors were rinsed in water following each analysis to prevent cross-contamination between samples.
2.6. Animals and Diets. Male C57BL/6J mice (3 weeks old) were purchased from Central Lab., Animal Inc. (Seoul, Korea). Animals were acclimatized for two weeks under a 12h:12h light-dark cycle and constant temperature (20 ± 2°C) and humidity (50 ± 5%), with water and food freely available. The mice were divided into 4 groups and then fed a normal diet containing 10% kcal fat (3.85 kcal/g) or a high-fat diet containing 60% kcal fat (5.24 kcal/g) (Research Diets Inc., New Brunswick, NJ). EALT dissolved in 0.5% CMC was orally administered (100 mg/kg body weight) once a day for 8 weeks. The high-fat diet mice received either 15 mg/kg body weight Orlistat (Xenical®, Roche Pharma Ltd., Reinach, Switzerland) or 0.5% CMC (10 ml/kg body weight) orally as a positive or negative control. A normal diet group was also treated with only 0.5% CMC (10 ml/kg body weight) as a vehicle. All animal experiments were carried out according to the guidelines of the Gyeongnam Department of Environment & Toxicology, Korea Institute of Toxicology, on the Care and Use of Laboratory Animals. Four groups are as follows.

Group I (ND) received oral 0.5%-CMC and was fed a normal diet (normal control group).

Group II (VC) received oral 0.5%-CMC and was fed a high-fat diet, used as a disease group (negative control group).

Group III (PC) received Orlistat (15 mg/kg body weight/rat, p.o.) and was fed a high-fat diet (positive control group).

Group IV (T1) received the low-temperature dried A. hookeri (100 mg/kg body weight/rat, p.o.) and was fed a high-fat diet.

Group IV (T2) received hot-air dried A. hookeri (100 mg/kg body weight/rat, p.o.) and was fed a high-fat diet.

2.7. Measurement of Body and Adipose Tissue Weight, Nonfasting Blood Glucose, and Blood Biochemistry. The body weight and blood glucose concentration were measured under the condition of nonfasting once a week. After 8 weeks of treatment, mice were sacrificed by cervical dislocation and the epididymal and perirenal adipose tissue were dissected and weighed. Nonfasting blood glucose level was measured by a standard method using a glucometer (Accu-Chek Active; Roche Applied Science, Indianapolis, IN). Blood samples at time 0. Two hundred µl of blood was sampled from tail at 15, 30, 60, and 120 minutes for both OGTT and IPITT. Blood glucose concentration was determined with a glucometer (Accu-Chek Active; Roche Applied Science, Indianapolis, IN).

2.9. Statistical Analysis. Each data value was presented as the mean ± SD. Data of body weight were analyzed by two-way ANOVA, while in vitro assay and in vivo biochemical parameters were analyzed by one-way ANOVA. The data were considered to be significant statistically if the probability had a value of 0.05 or less.

3. Results

3.1. Comparison of Low-Temperature and Hot-Air Drying Methods for the Amount of Volatile Sulfur Compounds from A. hookeri. Figure 1 presented the GC chromatographic profiles of the sulfur-containing volatile compounds in A. hookeri dried by low-temperature (Figure 1(a)) or hot-air method (Figure 1(b)). Peak identification for the chromatograms is given in Table 1 along with the organosulfurs content (mg/100 g) of the A. hookeri extracts. Major sulfur components contained in A. hookeri were identified as diallyl trisulfide, dimethyl trisulfide, dipropyl trisulfide, diallyl disulfide, and methyl allyl disulfide. In detail, diallyl trisulfide was the primary volatile (10.46 mg/100 g) in low-temperature-dried A. hookeri (T1), followed by dimethyl trisulfide (6.30 mg/100 g), dipropyl trisulfide (2.74 mg/100 g), and diallyl disulfide and methyl allyl disulfide (2.53 mg/100 g). The main organosulfur compounds were detected in decreased abundance in hot-air/long-time dried A. hookeri (T2) compared to T1. Low thermal treatment decreased the content of major sulfide compounds from T2, diallyl trisulfide (98.09% decrease), dimethyl trisulfide (85.56% decrease), dipropyl trisulfide (90.15% decrease), diallyl disulfide (100% decrease), and methyl allyl disulfide (98.81% decrease).

3.2. Effects of T1 and T2 in GC-MS-Olfactory and Sensory Tests. Principal information on the odor profile of T1 and T2 was obtained by utilizing a GC-MS-Sniff technique. As shown in Figure 2, the intensity of the olfactory peaks in T1 was markedly changed by hot-air and long-time thermal drying in T2. In comparison to T1 (Figure 2(a) and Table 2), most of the sulfur-containing compounds were detected at lower intensities in T2 (Figure 2(b) and Table 3). It was found that some important odorous sulfides such as di-2-propenyl trisulfide and dimethyl tetrasulfide were almost removed (>95%) in the aromagram of T2. The taste evaluation of T1 and T2 was performed with the electronic tongue data with six sensors including salty, sour, sweet, bitter, spicy, and umami (Figure 3). The spiciness and saltiness scores of T2 decreased to 42.11% and 50%, respectively, when compared to the T1 sample. In contrast, the sensory scores of sourness, sweetness, and umami taste of T2 were 100%,
3.3. Effects of T1 and T2 on Body Weight Gain, Adipose Tissue Weight, Nonfasting Blood Glucose Level, and Blood Chemistry in High-Fat Diet-Induced Mice. The body weight gain of mice fed a high-fat diet was greater than the value for the ND (normal diet) group (Figure 4). A 2.8-fold increase in body weight gain was observed in VC (high-fat diet + vehicle control) group compared with ND group. Orlistat is an antiobesity drug that is currently available as a strong gastrointestinal lipase inhibitor [23]. In the present study, Orlistat has been used as a positive control in which it effectively decreased weight gain by 26.96% ($P<0.01$) at the end of experiment (week 8). The average body weights of T1- and T2-administered mice (30.73 g and 29.46 g, resp.) were considerably lower than that of the VC group (32.31 g). The adipose tissue weight, including epididymal and perirenal fat, of obese mice was also markedly reduced by treatment of Orlistat and A. hookeri (Figure 5). Cumulative fat mass of high-fat diet group was estimated as the 4.5-fold value of nontreated mice. Treatment of Orlistat and T2 lowered the adipose tissue weight to 42.98% and 32.15%, respectively, when compared to the VC group. Nonfasting blood glucose concentrations of the five groups of mice were shown in Figure 6. High-fat diet feeding (197.63 mg/dl) increased in the nonfasting blood glucose level compared to the ND group (182.25 mg/dl) after seven weeks. T2 administration markedly prevented the elevation of blood glucose by 82.11% compared to VC group. However, the difference in blood glucose level between the T1 treatment group and VC group was not statistically significant. The remarkable change in blood biochemistry was the reduction in the level of triglyceride in A. hookeri-treated groups. The increased level of triglyceride in VC (127.06 mg) compared to ND (85.38 mg) was significantly decreased by the treatment with T1 (51.76 mg) or T2 (45.92 mg) that was more potent than positive control, Orlistat-treated group (65.98 mg) (Table 4).

### Table 1: Sulfur-containing volatile compounds in samples by GC/MS.

| RT (min) | Compounds                      | Content (mg/100 g) |
|---------|--------------------------------|--------------------|
|         |                                | T1                 | T2                 |
| 4.27    | Ethyl sulfide                  | —                  | 0.07               |
| 5.14    | Dimethyl sulfide               | 1.18               | 0.06               |
| 8.37    | 3,3-Thiobis-1-propene          | 0.07               | —                  |
| 8.90    | n-Propyl cis-1-propenyl sulfide| 0.06               | —                  |
| 10.58   | Methyl allyl disulfide         | 2.53               | 0.03               |
| 10.82   | Diethyl disulfide              | —                  | 0.02               |
| 11.17   | Methyl-trans-propenyl-disulfide| 1.32               | —                  |
| 11.22   | Methyl propyl disulfide        | —                  | 0.04               |
| 11.55   | Trans-propenyl methyl disulfide| —                  | 0.01               |
| 12.58   | Dimethyl trisulfide            | 6.30               | 0.91               |
| 17.17   | Propene sulfide                | 0.70               | —                  |
| 17.85   | Diallyl disulfide              | 2.53               | —                  |
| 18.28   | Butyl propenyl sulfide         | —                  | —                  |
| 18.83   | (E,E)-Bis(1-propenyl) disulfide| 2.33               | —                  |
| 19.18   | trans-Propenyl propyl disulfide| —                  | 0.03               |
| 19.51   | 2-Propenyl propyl disulfide    | 0.45               | 0.01               |
| 20.60   | Methyl methylthiomethyl disulfide| —            | —                  |
| 21.14   | Methyl propyl trisulfide       | —                  | 0.62               |
| 24.09   | Dimethyl tetrasulfide          | 1.32               | —                  |
| 27.91   | Diallyl trisulfide             | 10.46              | 0.20               |
| 28.94   | Dipropyl trisulfide            | 2.74               | 0.27               |
| 31.25   | Methyl 2-propenyl tetrasulfide | —                  | 0.23               |
| 34.39   | Dimethyl trisulfide            | —                  | 0.04               |
| 37.65   | Di-2-propenyl tetrasulfide     | 1.51               | 0.05               |

RT: retention time. “—” corresponds to “not detected.”

100%, and 85.71% higher than that of T1. The difference in bitter taste between the treatment groups was not statistically significant.
3.4. Effects of T1 and T2 on High-Fat Diet-Induced Glucose Intolerance and Insulin Resistance. OGTT and IPITT were performed to evaluate the effect of *A. hookeri* on glucose and insulin tolerance. Feeding a high-fat diet to mice for 8 weeks developed severe glucose intolerance and insulin resistance, characterized by increased area under the blood glucose curves (Figure 7). During OGTT, the blood glucose of high-fat diet-fed mice increased to 498.38 mg/dl after 30 min from a baseline value of 248.75 mg/dl (Figure 7(a)). The blood glucose levels of T1 and T2 group decreased to 306.25 mg/dl and 395.38 mg/dl, respectively, at 30 min. In IPITT, mice treated with a high-fat diet also exhibited severe insulin resistance (Figure 7(b)). In mice treated with T1 and T2 following i.p. insulin load, the level of blood glucose decreased to 110.75 mg/dl and 131.50 mg/dl, respectively, after 30 min in comparison to the vehicle control (163.88 mg/dl). Blood glucose levels of PC, a positive control, were 395.13 for OGTT and 96.63 for IPITT after 30 min, which were 20.72% and 41.04% less as compared to corresponding values observed in high-fat diet control mice.

4. Discussion

Drying is an ancient process used to preserve botanical medicines. Drying process is based on the dehydration of plants to extend the shelf-life and improve the nutritional value [22, 24]. Various drying techniques, hot-air, microwave, and freeze drying, and operating conditions have been applied in the preparation of botanical samples [25, 26]. Herbal extracts might vary depending on different operating conditions including temperature, relative humidity, and air velocity even in the same drying process [27]. Drying has significant effects on the production of quality medicinal plants. In addition, the constituent profiles of plants can be affected by the choice of drying method and thereby adoption of proper drying technique is important [26].

To compare the relative composition of volatile sulfur compounds, *A. hookeri* samples were prepared by two different drying methods, conventional low-temperature drying system minimizing the loss of volatile sulfur compounds and the optimized hot-air/long-time drying system minimizing unpleasant odor and taste. The characteristic flavor of *Allium*
Table 2: Odorous compounds in T1 by GC/MS-olfactometry.

| RT (min) | Compounds                                      | Content (ug/100 g) |
|---------|------------------------------------------------|--------------------|
| 6.93    | Dimethyl disulfide                             | 0.11               |
| 8.49    | 2,3,3-Trimethyl-1,4-pentadiene                 | 0.02               |
| 8.61    | Hexanal                                        | 0.07               |
| 9.66    | 5-Methyl-2-thiophenecarboxaldehyde             | 0.14               |
| 9.97    | 6,6-Dimethylhepta-2,4-diene                    | 0.09               |
| 10.16   | 1,6-Dimethyl-1,5-cyclooctadiene                | 0.07               |
| 10.27   | 1,2,4,4-Tetramethylcyclopentene                | 0.07               |
| 10.67   | Ethyl-benzene                                  | 0.17               |
| 11.63   | 1,2-Bis(allyl)disulfane                        | 2.73               |
| 12.17   | 3-(Methylthio)-propanal                        | 1.72               |
| 13.00   | Methyl propyl disulfide                        | 1.06               |
| 13.63   | Methyl (methylthio)methyl disulfide            | 0.90               |
| 14.32   | Diethyl trisulfide                             | 5.22               |
| 14.98   | 2-Pentyl-furan                                 | 0.24               |
| 15.51   | Cyclodecanone                                  | 0.34               |
| 16.70   | Benzenecacetaldyhyde                           | 0.11               |
| 17.02   | Dodecane                                       | 0.27               |
| 17.19   | 2-Formyl-3-methylthiophene                     | 0.09               |
| 17.90   | Di-2-propenyl disulfide                        | 0.34               |
| 18.44   | Dimethyl tetrasulfide                          | 10.81              |
| 20.42   | 1,2-Dithiane-4,5-diol                          | 1.38               |
| 21.78   | 3-Vinyl-1,2-dithiacyclohex-5-ene               | 0.15               |
| 22.19   | Di-2-propenyl trisulfide                       | 13.73              |
| 22.62   | 2,4-Dihydro-4-methyl-3H-1,2,4-triazole-3-thione| 6.91               |
| 22.80   | 1-Ethoxyvinyl chloroacetate                    | 0.27               |
| 24.22   | Di-2-propenyl trisulfide                       | 13.94              |
| 24.53   | 2,4-Dihydro-4-methyl3H-1,2,4-triazole-3-thione  | 7.04               |
| 26.34   | Methyl 2-propenyl disulfide                    | 1.97               |

RT: retention time.

Table 3: Odorous compounds in T2 by GC/MS-olfactometry.

| RT (min) | Compounds                                      | Content (ug/100 g) |
|---------|------------------------------------------------|--------------------|
| 9.72    | 2-Furancarboxaldehyde                          | 1.05               |
| 10.55   | 2-Furanmethanol                                | 0.08               |
| 11.73   | Pentanoic acid                                 | 0.05               |
| 12.38   | 1-(2-Furanyl)-ethanone                         | 0.05               |
| 13.00   | Methyl propyl disulfide                         | 0.04               |
| 14.11   | 5-Methyl-2-furancarboxaldehyde                 | 0.09               |
| 14.29   | Dimethyl trisulfide                             | 1.03               |
| 15.71   | Hexanoic acid                                  | 0.09               |
| 16.68   | 1H-Pyrrole-2-carboxaldehyde                    | 0.17               |
| 16.79   | Benzenecacetaldyhyde                           | 0.02               |
| 17.01   | Dihydro-p-toluinaldehyde                       | 0.05               |
| 17.80   | Heptanoic acid                                 | 0.06               |
| 18.36   | 2,6,11,15-Tetramethyl-hexadecane               | 0.02               |
| 18.46   | Nonanal                                        | 0.05               |
| 18.88   | Phenylethyl alcohol                            | 0.07               |
| 19.08   | 2-Ethyl-hexanoic acid                          | 0.13               |
| 20.63   | Octanoic acid                                  | 0.13               |
| 21.87   | Dimethyl tetrasulfide                          | 1.03               |
| 23.39   | Tetradecane                                    | 0.08               |
| 24.12   | Di-2-propenyl trisulfide                       | 0.44               |
| 24.80   | Dipropyl trisulfide                            | 0.65               |
| 26.29   | Methyl 2-propenyl disulfide                    | 0.40               |

RT: retention time.

Table 4: Blood biochemistry in high-fat diet-induced obese mice.

|        | HDLC | LDLC | TC  | TG  |
|--------|------|------|-----|-----|
| ND     | 62.54| 5.84 | 95.80| 85.38|
| VC     | 81.74| 5.70 | 135.00| 127.06|
| PC     | 75.34| 8.44 | 130.00| 65.98|
| T1     | 81.24| 7.02 | 142.80| 51.76|
| T2     | 79.04| 7.38 | 142.00| 45.92|

1 HDLC: high-density lipoprotein cholesterol.  
2 LDLC: low-density lipoprotein cholesterol.  
3 TC: total cholesterol.  
4 TG: triglyceride.

dimethyl trisulfide, dipropyl trisulfide, diallyl disulfide, and methyl allyl) was significantly lower in *A. hookeri* dried with hot-air (T2) than in *A. hookeri* dried at low-temperature (T1). Furthermore, some sulfur compounds of T1 such as diallyl disulfide, dimethyl tetrasulfide, (E,E)-bis(1-propenyl) disulfide, and methyl-trans-propenyl-disulfide did not exist in T2 sample. According to the previous literatures, high temperature convective drying normally resulted in poor sulfur compounds retention and moderate air temperatures allowed a better sulfur compounds content than harsh thermal treatment [5, 6]. Consistent with the report, in our present study, drastic qualitative and quantitative reduction of the major sulfide compounds was observed after hot-air thermal drying.
Figure 3: Consumer acceptance of T1 and T2. Blue line represents T1 (low-temperature dried *A. hookeri*) and red line represents T2 (hot-air dried *A. hookeri*).

Figure 4: Effects of T1 and T2 on body weight of obese mice fed a high-fat diet for 8 weeks. Values were expressed as the means ± SD (n = 10). ND, normal diet group; VC, high-fat diet group; PC, Orlistat-treated group; T1, low-temperature dried *A. hookeri*-treated group; T2, hot-air dried *A. hookeri*-treated group.

Figure 5: Effect of T1 and T2 on adipose tissue weights in high-fat diet-induced obese mice. Values were expressed as the means ± SD (n = 10). ND, normal diet group; VC, high-fat diet group; PC, Orlistat-treated group; T1, low-temperature dried *A. hookeri*-treated group; T2, hot-air dried *A. hookeri*-treated group.

Figure 6: Effect of T1 and T2 on nonfasting blood glucose. Values were expressed as the means ± SD (n = 10). ND, normal diet group; VC, high-fat diet group; PC, Orlistat-treated group; T1, low-temperature dried *A. hookeri*-treated group; T2, hot-air dried *A. hookeri*-treated group.
Figure 7: Effect of EALT on OGTT (a) and IPITT (b). Values were expressed as the means ± SD (n = 10). ND, normal diet group; VC, high-fat diet group; PC, Orlistat-treated group; T1, low-temperature dried *A. hookeri*-treated group; T2, hot-air dried *A. hookeri*-treated group.

The GC/MS data of T1 and T2 were correlated with the result of their GC/MS-olfactometry. Comparison of the chromatographic odor profiles between T1 and T2 showed that the odor intensity of most of the olfactory peaks was definitely weakened in T2. To verify the reduced spiciness of T2, actual score of spicy taste was determined by human sensory testing using an electronic tongue. In comparison to the T1, T2 treatment revealed a significant decrease in spiciness and saltiness but increase in sweetness and umami taste. Spicy odor and flavor of *Allium* plants are known to be occur as a result of generation of major sensory-active sulfur compounds [29]. In general, the tastes of sweet and umami cause food acceptance behavior, whereas the bitter and salty tastes elicit avoidance [30]. According to overall GC/MS-olfactometry and sensory test results, T2 exhibited better removal of taste- and odor-causing compounds and thereby the optimized process of drying with hot-air might encourage acceptance and consumption of *A. hookeri* products.

Antiobesity and hypolipidemic effects of *A. hookeri* were investigated in dietary obese mice. It has been reported that animal model of obesity induced by high-fat diet resembles the human obesity [31]. Adipocyte hypertrophy, hyperplasia, and insulin resistance, similar to those in human obese subjects, were observed in rodents after high-fat diet feeding [31]. In our present study, as expected, exposure of mice to a high-fat diet caused severe obesity characterized by increased body weight and adiposity. A significant reduction in body weight gain and fat deposition in obese mice was detected by treatment of *A. hookeri*. Particularly, the antiobesity effect was more pronounced with T2 in comparison to T1. According to histological analysis of epididymal fat, adipocytes from VC group were markedly larger than those from ND group. An increase in total white adipose tissue mass of epididymal and perirenal fat under high-fat diet was inhibited in 100 mg/kg T2-administered mice. Sulfur-containing compounds, such as allicin and allyl methyl sulfide, are known to be the main responsible constituents of *Allium* plants for most of their pharmacological effects [32, 33]. Although total amount of volatile sulfur compounds tended to be lower in T2 than in T1, T2 showed better activity for decreasing weight gain and adiposity in high-fat diet-induced animal model.

Consumption of high levels of dietary fat is considered to be a major factor in the promotion of hyperglycemia and whole-body insulin resistance [34]. An elevated glucose concentration is a common feature of obesity itself or closely linked metabolic diseases [31]. In this regard, high-fat diet-induced obesity is directly connected with an increased risk of type 2 diabetes [35]. Our observations of hyperglycemia and insulin tolerance in mice fed high-fat diet confirmed induction of dietary obesity. The increased level of nonfasting blood glucose induced by high-fat diet was attenuated by T2 treatment for seven weeks. Besides, T2 coadministration with high-fat diet prevented both glucose and insulin tolerance during OGTT and IPITT. Taken together, T2 might contribute to improvement of insulin sensitivity and resultant lowering of the elevated blood glucose level in obesity in spite of its reduction of sulfur-containing components.
5. Conclusion

Collectively, the content of major sulfide compounds was noticeably diminished by the optimized thermal drying with hot-air and thereby a remarkable decrease in spiciness and odorous intensities was observed in T2. Administration of T2 effectively prevented weight gain and decreased body fat mass induced by high-fat diet feeding in animal model, although bioactive sulfur-containing compounds were decreased in T2. T2 was also useful in improving blood chemistry and mitigating both glucose tolerance and insulin tolerance in dietary obese mice. Further study is warranted to explore the active components responsible for the antiobesity effect of T2.

Ethical Approval

The project was conducted under the approval of Uganda National Council for Science and Technology (Reference no. HSI3313) and Makerere University College of Health Sciences Higher Degree Research and Ethics Committee (Reference no. 2011-145).

Competing Interests

The authors declare that there are no competing interests regarding the publication of this paper.

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