OH terminated two-dimensional transition metal carbides and nitrides (MXenes) as ultralow work function materials

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Abstract: MXenes are a set of two-dimensional transition metal carbides and nitrides that offer many potential applications in energy storage and electronic devices. As an important parameter to design new electronic devices, we investigate the work functions of bare MXenes and their functionalized ones with F, OH, and O chemical groups using first-principles calculations. From our calculations, it turns out that the OH terminated MXenes attain ultralow work functions between 1.6 and 2.8 eV. Moreover, depending on the type of the transition metal, the F or O functionalization affects increasing or decreasing the work functions. We show that the changes in the work functions upon functionalizations are linearly correlated with the changes in the surface dipole moments. It is shown that the work functions of the F or O terminated MXenes are controlled by two factors: the induced dipole moments by the charge transfers between F/O and the substrate, and the changes in the total surface dipole moments caused by surface relaxation upon the functionalization. However, in the cases of the OH terminated MXenes, in addition to these two factors, the intrinsic dipole moments of the OH groups play an important role in determining the total dipole moments and consequently justify their ultralow work functions.
1. Introduction

The work function is one of the basic physical characteristics of a material that is related to the bulk properties (through the Fermi energy) and is influenced by the surface properties such as the surface index, the surface relaxation, and the surface adsorbates [1-11]. Moreover, the work function is one of the most important technological parameters in developing new field emitter cathodes or new Schottky barrier junctions for the applications in the light emitting diodes and the field effect transistors [12,13]. Materials with low work functions are demanded for many electronic applications so as to enhance the power efficiency. A typical example of such materials is cesium, for which the work function is 2.1 eV [1]. It is also known that the cesium plays a crucial role in lowering the work function when it is deposited on various surfaces [14-16]. This is along the general ideas that the deposition of the alkali metals or other elements is capable of tuning the surface dipole moments, which in turn directly affect the work functions [17-22]. However, the usage of the cesium is always accompanied with some major concerns because of its toxicity. Recently, new techniques have been developed successfully to grow self-assembled monolayers of organic molecules with low work functions, which are eco-friendly [23] although they might be unstable thermally.

Various metallic substrates of transition metal carbides and nitrides provide the high stabilities, the high melting points [24], and the relatively low work functions [25-28] that have increasingly attracted attentions as suitable field emitters [29,30]. Their thin films also have already been synthesized experimentally [30]. Remarkably, experimentalists succeeded recently to synthesize the two-dimension monolayers and multilayers of a particular family of transition metal carbides and nitrides with chemical composition of $M_{n+1}C_n$ and $M_{n+1}N_n$ ($M$ is an early transition metal), which were named MXenes [31,32]. The MXenes are derived from exfoliation of the MAX phases [33-37] --- $M_{n+1}AX_n$, where A is an element from groups 13-16 in the periodic table and X is carbon or nitrogen --- using the hydrofluoric acid. During the acid treatment, the A element is removed from the MAX phase and the surfaces of the obtained two-dimensional $M_{n+1}X_n$ are instantaneously terminated by F, OH and/or O groups. The MXenes have already found applications as energy storage materials in Li ion batteries and high volumetric capacitors [38-46], as transparent conductive
electrodes [47], filler in polymeric composites [48], purifier [49], dual-responsive surfaces [50], and a suitable substrate for dyes [51]. Theoretically, there are many proposed applications for the monolayers, multilayers, and heterostructures of the MXenes in the electronic [52-64], magnetic [65,66], optoelectronic [67], thermoelectrics [52,68], sensors [69], Schottky barrier junction devices [70-72].

Considering properties of the MXenes, it is noticed that the members of this family can be good candidates as materials with the low work functions by the following reasons: first, the MXenes offer the tunability of the work function by choosing the proper transition metal as well as the X element, usually carbon or nitrogen. In addition, such compositional tunability provides a possibility to adjust other properties such as thermal, mechanical, chemical stabilities, and toxicity. Second, during the synthesis of the MXenes, their surfaces are functionalized naturally, affecting the electronic structure, especially, giving the Fermi level shift and the electrostatic potential change near the surfaces. They are caused by the electron redistribution in energy and in real space, respectively. Both changes have significant effects on the work function. Currently, the MXenes are functionalized by mixture of O, F, and/or OH. From a viewpoint of the development of low work function materials, the OH termination is interesting because the OH group has an intrinsic dipole moment.

In this paper, using first-principles calculations based on the density functional theory (DFT), we propose the functionalized MXenes as materials with the tunable low work functions. Here, we have studied the work functions of a large number of the two-dimensional MXenes with various thicknesses and transition metals such as $\text{M}_2\text{C}$ and $\text{M}_{10}\text{C}_9$ ($\text{M}=\text{Sc, Ti, Zr, Hf, V, Nb, and Ta}$) as well as $\text{M'}_2\text{N}$ and $\text{M'}_{10}\text{N}_9$ ($\text{M'}=\text{Ti, Zr, and Hf}$) functionalized with F, OH, and O. As examples of structures of the functionalized MXenes, the model structures of $\text{M}_2\text{C(OH)}_2$ and $\text{M}_{10}\text{C}_9(\text{OH})_2$ are shown in Fig. 1. It is predicted from the calculations that the OH terminated carbide or nitride MXenes attain the ultralow work functions between 1.6 and 2.8 eV comparable with the cesium-deposited surfaces. In addition, depending on the M element, the F or O termination affects the work function to increase or decrease as compared with their corresponding bare structures without functionalization.
2. Calculation and analysis method:

We used DFT with the Perdew–Burke–Ernzerhof (PBE) version of the generalized gradient approximation (GGA) for the exchange-correlation functional [73] to optimize atomic structures and calculate electronic structures. The basis sets were generated using the projector augmented wave (PAW) method with a cutoff energy of 520 eV. The Methfessel–Paxton smearing scheme was applied with a smearing width of 0.1 eV [74]. The two-dimensional system was modeled by using the supercell approximation with a vacuum space along the direction perpendicular to the surface, which is taken to be the z-axis. The atomic positions and the two-dimensional cell parameters were fully optimized using the conjugate gradient method so that the magnitude of the force acting on each atom became less than 0.001 eVÅ−1 in the optimized structures and the total energies were converged within $10^{-7}$ eV/cell. For the structural optimization, the Brillouin zone was sampled using a set of $12 \times 12 \times 1$ k points [75]. All calculations were performed using the VASP code [76].

As described above, the exposed surfaces of the MXenes are functionalized with F, OH, and/or O during exfoliation process. We have previously shown that the MXenes with the full surface functionalizations are thermodynamically more favorable than the partial functionalizations [52], where the full surface functionalization requires two chemical groups per cell. The surfaces of the MXenes have two types of the hollow sites: hollows with and without a carbon/nitrogen sites beneath them [52]. Thereby, we tested three initial configurations to find the optimum positions for the attached chemical groups: Structure 1) both chemical groups are attached to hollows without carbon/nitrogen; Structure 2) both chemical groups are attached to hollows with a carbon/nitrogen atom; Structure 3) two chemical groups are attached to two different types of hollows. The optimization results of the above three initial configurations are listed in the supplemental material [77]. We used the lowest energy configurations for the work function analyses. Moreover, for the cases of the OH-terminated systems, we performed additional test calculations by tilting the H atoms toward the nearest transition metals and carbon/nitrogen atoms. It was found in the considered MXenes the OH groups locate always perpendicular to the surfaces. We also calculated the phonon dispersions of some of the OH-terminated MXenes by using Phonopy package [78]. As shown below, the phonon modes are all positive, indicating the stability of such systems.
The work function was derived from the energy difference between the Fermi energy and the vacuum level (electrostatic potential away from the surface). We used an extremely large vacuum space of 80 Å to prevent the interactions between monolayers and their images along the z-axis. The convergence error of the calculated work function is smaller than 1.0 meV (see the supplemental material [77]). The structures are placed in the center of the cell. We validate our computational approach by performing sets of work function calculations on pure transition metals as well as binary transition metal carbides and nitrides [77]. The results were compared with the available theoretical and experimental data [79-91].

By using the Poisson equation, the electrostatic potential away from the surface can be related to the surface dipole moment, and thus, the work function [92]. Leung et al. derived an expression relating the change in the work function (ΔΦ) and the change in the surface dipole moment (ΔP): ΔΦ = -e/εoΔP = -180.95ΔP, where ΔΦ = Φ-Φo, ΔP = p-po, and e (C) and εo (CV-1 Å-1) are the charge of electron and vacuum permittivity, respectively [92]. The Φo (Φ) and po (p) are the work function and the surface dipole moment before (after) functionalization, respectively. In order to obtain the surface dipole moments, we used the average charge density (ρ(z)) along the z-axis, which was obtained from the total charge density ρ(x,y,z) by ρ(z) = (1/A) ∫∫ dx dy ρ(x,y,z) where A is the cell surface area. It is noted that the total charge density includes the contributions of both electrons and ions. The ion charge densities are defined as point charges using the delta function distributions at the positions of the ions. The surface dipole moment can be determined by p(z) = ∫z0 z′ρ(z′)dz′, where zo (0.0 Å) is the center of the MXene.

Upon deposition of any element, three main factors control the surface dipole moment: i) the redistribution of electron charge between the surface and the adsorbates, ii) the surface relaxation caused by the adsorbates, and iii) the polarity of the adsorbates. Following the theoretical method developed by Leung et al. [92], the electron charge transfer (Δρ(z)) between the adsorbates and the substrate is defined as Δρ(z)=ρ(z)-[ρs(z)+ρa(z)], where ρ(z), ρs(z), and ρa(z) are the total charge densities of the functionalized system, the substrate, and the adsorbate, respectively. Computationally, ρs(z) is generated from a total energy calculation of the substrate by removing the overlayer of adsorbates from the functionalized system without the structural optimization. Similarly, ρa(z) is obtained from a total energy calculation of
the adsorbate layer by removing the substrate from the functionalized system. Thereby, the change in the total dipole moment (ΔP) by functionalization can be expressed as ΔP = Δp + p_s - p_o, where Δp, p_s, and p_o are the induced dipole moments of the Δρ(z), ρ_s(z) and ρ_o(z), respectively. As a consequence, the change in the work function becomes ΔΦ = -(e/ε_o)(Δp + p_s - p_o). Therefore, ΔΦ is affected by the dipole moment of the transferred charges (Δp) between the substrate and the adsorbates, the dipole moment of the adsorbates (p_a), and the change in the dipole moments due to substrate relaxation (p_s - p_o).

The work function of a functionalized two-dimensional system can be affected by another factor, the shift of the Fermi level. In contrast to the three-dimensional systems, the Fermi energy can shift by the electron transfer caused by the surface functionalization. Therefore, the surface functionalization of thin materials can change both the Fermi energy and the surface dipole moments, but in the bulk three-dimensional materials, the surface functionalization modifies mainly the surface dipole moments.

3. Results and discussions:

We have systematically studied the work functions of the MXenes with different M elements: M_2C and M_{10}C_9 (M= Sc, Ti, Zr, Hf, V, Nb, Ta) as well as M'_{2}N and M'_{10}N_9 (M'= Ti, Zr, and Hf) functionalized with F, OH, and O. The results of the calculated work functions are shown in Fig. 2. Figure 2 does not include the results for Sc_2CO_2, Ti_2NF_2, and Ti_{10}N_9F_2, in which the functional groups are adsorbed on different types of the hollow sites. These MXenes obtain intrinsic dipole moments due to the asymmetric attachments of the functional groups. Here, we focus only on the MXenes without the intrinsic dipole moments to understand the systematical change of the work functions.

Figure 2 clearly shows that in each MXene family, their thin and thick monolayers manifest similar behaviors upon the same functionalization. The work functions of the bare MXenes and the functionalized ones by F, OH and O are found to be distributed in the range of 3.3-4.8, 3.1-5.8, 1.6-2.8, and 3.3-6.7 eV, respectively. Specially, we found that that M_2C(OH)_2, M_{10}C_9(OH)_2, M'_{2}N(OH)_2 and M'_{10}N_9(OH)_2 exhibit ultralow work functions as compared with their corresponding bare and the functionalized ones with F or O. Among these systems, Sc_2C(OH)_2 and Sc_{10}C_9(OH)_2
exhibit the lowest work functions of ~1.6 eV. However, depending on the M element, the work function of the functionalized MXenes with F or O can be higher or lower than that of their corresponding bare structures. For instance, the F functionalization on Sc_{10}C_9 increases the work function up to 4.99 eV from the bare value 3.35 eV, but on Hf_{10}C_9 decreases the work function down to 3.19 eV from the bare value 4.76 eV. The O functionalization increases the work functions for most of the considered MXenes, except for Hf_{10}N_9 and Sc_{10}C_9. Among the considered MXenes, V_2CO_2 and V_{10}C_9O_2 possess the highest work functions of ~6.7 eV. The work function reduction upon the F or O functionalization is apparently in contradiction with the common accepted rule that high electronegative elements such as F or O increase the work functions. It is worth mentioning that such unexpected behavior has also been reported in other materials. For example, oxygen adsorption on the tungsten (100) shows the work function reduction [92].

It is noteworthy that MAX phases with n ≥ 3 exist: Ti_4SiC_3 [93], Ti_4AlN_3 [94], Ti_6GeC_3 [95], V_4AlC_3 [96], Nb_4AlC_3 [97], Ta_4AlC_3 [98], Ta_5GaC_3 [99], Ti_5SiC_4 [93], (Ti_{0.5}Nb_{0.5})_3AlC_4 [100], Ta_6AlC_5 [101], Ti_7SnC_6 [102]. These experimental observations indicate that formations of MAX phases (M_{n+1}AX_n) with n ≥ 3 are possible though the production of their single phases may need further optimization of the experimental conditions. Thereby, it is fundamentally interesting to study the thickness dependence of the work functions of the derived M_{n+1}X_n MXenes in detail. Hence, next we investigated the thickness dependence of the work function functionalized by F, OH, and O. As typical examples, we calculated the work functions of Ti_{n+1}C_n, Ti_{n+1}C_nF_2, Ti_{n+1}C_n(OH)_2, and Ti_{n+1}C_nO_2 (n = 1-9) and the results are shown in Fig. 3. In these Ti-based MXenes, the thicknesses of the bare ones extend from 2.31 (Ti_2C) to 22.15 Å (Ti_{10}C_9). As seen in Fig. 3, the thin and thick MXenes act similarly upon the same functionalization: the F or O functionalization increases the work functions, while the OH functionalization decreases it significantly. In general, the dependency of work functions of the MXenes on their thicknesses is weak.

The calculations also indicate that the thickness affects the work functions of the thin MXenes. This is due to the quantum size effects [7-9,103], which result in the shift of the Fermi energy, and consequently the change of the work function. In very thin MXenes, few energy bands cross the Fermi energy. Therefore, any elemental deposition may have significant effects on the electronic structure, i.e., the number of
bands and band shapes, and thus the position of the Fermi energy varies (see the supplemental material [77]). The largest change in the work functions of the functionalized Ti_{n+1}C_n occurs between Ti_2CO_2 and Ti_3C_2O_2. This change is induced because of the specific properties of these systems: Ti_2CO_2 is a semiconductor [52] while Ti_3C_2O_2 is metallic (see supplemental material [77]). From thickness dependence analyses of the work functions of functionalized Ti_{n+1}C_n, it is recognized that the change of the Fermi energy, which in principle affects the value of the work functions, cannot explain the ultralow work functions found in the OH terminated Ti_{n+1}C_n.

Since the modifications of the work function of the MXenes upon different functionalizations cannot be justified by the change of the Fermi energy alone, it is necessary to examine the other contributions to the work function, i.e., the electrostatic potential. As described above the electrostatic potential away from the surface can be estimated from the surface dipole moment [92]. In the thick MXenes (M_{10}C_9 and M’_{10}N_9), the functionalization does not change the Fermi energy significantly. Hence, this allows us to focus on the electrostatic term only (surface dipole moments) to explain the changes in the work functions after functionalization. Figure 4 summarizes the results for the changes in the work functions (ΔΦ) of M_{10}C_9 and M’_{10}N_9 and their corresponding surface dipole moments (ΔP) upon different functionalization relative to their bare systems. In Fig. 4, the positive (negative) values of ΔΦ correspond to the increase (decrease) of the work function. It is found that ΔΦ correlates with -ΔP almost linearly with the slope value of 170.012 VÅ. The analysis by Leung et al [92] has shown that -ΔΦ/ΔP = 180.95 VÅ for the bulk surfaces, and no deviation were observed. In contrast, our results apparently deviate from this relationship. In their analysis, the shift of the Fermi level upon the functionalization was not taken into account because such shift does not occur in the bulk. On the other hand, the functionalization of monolayers can have a non-negligible effect on the Fermi level. Thus, the deviation is due to the finite shift of the Fermi level.

In order to understand how the chemical groups affect the surface dipole moments, we considered in details all terms of Δp, p_a, p_s, and p_o contributing to the surface dipole moments (ΔP = Δp + p_a + p_s - p_o) in details. As typical examples, we dissolve the total dipole moments into contribution of Δp, p_a, p_s, and p_o for systems, Hf_{10}C_9F_2, Sc_{10}C_9F_2, and Sc_{10}C_9(OH)_2. Note that the F functionalization decreases
increases) the work function of Hf$_{10}$C$_9$ (Sc$_{10}$C$_9$), while the OH functionalization decreases work function of Sc$_{10}$C$_9$. The results of $\Delta p$, $p_a$, $p_s$, and $p_o$ for these systems are shown in Figs. 5-7.

Figures 5-7 also include the redistributions of the transferred electron charges between the substrate and the functional groups ($\Delta \rho$). A common feature is that the charge redistributions are mainly localized in the regions nearby the surfaces and the adsorbates. It is also observed that in all MXenes studied here, the natures of F and O elements are the same: due to their higher electronegativity than transition metals and/or hydrogen atoms of OH, these elements always receive electrons from the nearby elements. Thereby, as shown in Figs. 5-7(a), several regions with positive (nearby transition metals and/or hydrogen indicated by ① and ③) or negative charges (nearby the F and O indicated by ②) are formed. The positive or negative charges in regions ①, ② and ③ contribute largely to the $\Delta \rho$ [92]. As shown in Figs. 5(b) and 6(b), the induced dipole moments $\Delta p$ of the redistributed charges $\Delta \rho$ for Hf$_{10}$C$_9$F$_2$ and Sc$_{10}$C$_9$F$_2$ becomes positive for Hf$_{10}$C$_9$F$_2$ at the distance of 40 Å from the surface while it is negative for Sc$_{10}$C$_9$F$_2$ at the same position. This clarifies why the F deposition causes the decrease of the work function in Hf$_{10}$C$_9$F$_2$, but the increase in Sc$_{10}$C$_9$F$_2$. In fact, the charge transfer has already been attributed to the increase or decrease of the work functions in many materials. For example, the charge transfer explains the effects of different adsorbates on the work functions of graphene or boron-nitride sheets [104-106], and the reduction of the work function for the tungsten (100) surface upon the oxygen deposition [92]. However, the low work function behaviors of the OH terminated MXenes cannot be explained by the effect of $\Delta \rho$ only. In addition to $\Delta \rho$, it is necessary to take into account the contributions of the induced dipole moments from surface relaxations ($p_s-p_o$) and the adsorbates ($p_a$) in the total dipole moments ($\Delta P$).

The term $p_s-p_o$ indicates the modifications in the surface dipole moments due to the changes in the atomic layer distances upon the chemical functionalization, i.e., the structural relaxation effect. In Figs. 5-7 (c), the sharp changes in the $p_s$ or $p_o$ occur at the position of the atomic layers. It is noted in Figs. 5-7 that in order to avoid the complications of the figures, we have shown only the position of the atomic layers in the functionalized MXenes but not in the bare MXenes. As seen in Fig. 5(c) the F functionalization on Hf$_{10}$C$_9$ does not change the atomic layer distances, and consequently $p_s-p_o$ is zero at the distance of 40 Å far away from the surface. However,
as seen from Figs. 6(c) and 7(c), upon the F or OH functionalization on Sc$_{10}$C$_9$, the layer distances near the surfaces change outwardly as compared with the bare surface. Hence, the term $p_s$-$p_o$ contributes to the total dipole moments. The reason why the F or OH adsorbates affect the layer distances of Sc$_{10}$C$_9$, but not in Hf$_{10}$C$_9$, might be explained by the layer elastic constants ($C_{33}$) as our $C_{33}$ calculations indicate that Sc$_{10}$C$_9$ (219.9 GPa) is softer than the Hf$_{10}$C$_9$ (457.7 GPa) [77].

Figures 5-7(d) show the dipole moment contributions of the adsorbate layers ($p_a$). As expected and also shown in Figs. 5(d) and 6(d), the layers of F or O groups do not have any net dipole moment away from the surface. However, OH is a polar chemical group. Hence, the OH layer has an intrinsic dipole moment as shown in Fig. 7(d). We found that the contribution of the OH dipole moment to the total dipole moments is large and its sign is opposite to the contributions of the other terms ($\Delta p$ and $p_s$-$p_o$). This explains why the OH terminated MXenes exhibit the exceptionally low work functions, in spite of the fact that OH has an electronegative nature similar to F and O groups. It should be noted that the importance of the effect of the intrinsic dipole moments of chemical groups has been pointed out on the increase or decrease of the work functions of the Au (111) surfaces as well. For example, it has been shown that the deposition of alkylthiolates (-SCH$_3$) increases the work function of Au (111), whereas deposition of fluorinated alkylthiolates (-SCF$_3$) decreases it [107].

In order to show the OH terminated MXenes are locally stable structures, we have performed a set of phonon calculations on Sc$_2$C(OH)$_2$, Ti$_2$C(OH)$_2$, Zr$_2$C(OH)$_2$, Zr$_2$N(OH)$_2$, Hf$_2$C(OH)$_2$, Hf$_2$N(OH)$_2$, and Ta$_2$C(OH)$_2$. The phonon modes of these systems are found to be all real and positive. The phonon dispersions of Sc$_2$C(OH)$_2$ (the MXene with lowest work function) and Ti$_2$C(OH)$_2$ (one of the well studied MXenes experimentally [31,32,38,47,108,109]) are shown in Fig. 8. The others are summarized in the supplemental material [77]. Interestingly, in Fig. 8 the phonon branches above 100 THz correspond to the stretching modes of OH bonds [109]. The weak dispersion indicates that the interaction between the adjacent OH groups is small. The OH frequencies of MXenes (see the supplemental data [77]) are comparable to the OH stretching modes of H$_2$O (3585.5 cm$^{-1}$) or alcohol (3500-3700 cm$^{-1}$). This means that the OH bonds are fairly strong. These evidences imply that the hydrogen atoms may not be easily detached from the OH terminated MXenes. Moreover, recently experimentalists have successfully studied the OH stretching frequency modes (~3746 cm$^{-1}$) on silica at different temperatures up to 500 K [110].
The OH vibrational modes of MXenes are in the same range as that of silica [77]. Since the thermal stabilities of ceramics are typically higher than silica, it is expected the OH terminated MXenes to be also thermally stable at high temperatures around 500 K.

In order to compare the work function properties of the MXenes with other related materials, we have performed sets of calculations on the binary carbides (MC, M= Sc, Ti, Zr, Hf, V, Nb, Ta) and nitrides (M’N, M’= Ti, Zr, and Hf) as well as pure transition metals (Sc, Ti, Zr, Hf, V, Nb, Ta) functionalized with F, O, and OH. The results are summarized in the supplemental data [77]. We found only one experimental result on TiC that after oxygen exposure, its work function increases [111]. Such increase in the work function is also seen from our results. From our calculations, it is observed that in most of the MC and M’N systems, the F, O, or OH functionalization causes an increase in the work functions. Interestingly, the OH terminations on MC and M’N do not help to lower the work functions significantly. This is because the OH groups on MC and M’N are tilted. Hence, the polarity of the OH groups does not affect to lower the effective potentials and consequently the work functions of the MC and M’N systems. The work function properties of the functionalized transition metal surfaces are very similar to those for the functionalized MXenes [77]. In these systems, the OH groups stay perpendicular to the surface similar to the OH groups on MXene structures. Hence, the OH terminated transition metal surfaces obtain ultralow work functions similar to the OH terminated MXenes. However, owing to the ceramic nature of the MXenes, MXenes possess higher thermal stabilities than the pure transition metals and are superior for work function applications.

4. Summary:

By using a set of first-principles calculations, we studied large numbers of thin and thick MXenes: M₂C and M₁₀C₉ (M= Sc, Ti, Zr, Hf, V, Nb, and Ta), as well as M’₂N and M’₁₀N₉ (M’= Ti, Zr, and Hf) functionalized with F, OH, and O. Our calculations revealed that independently of the type of M elements, the OH terminated MXenes attain ultralow work functions. In addition, it was observed that the increase or decrease in the work functions of the MXenes upon F or O functionalization depends on the type of the M elements. Moreover, we systematically studied the thickness
dependences of the work functions of the functionalized Ti\(_{n+1}C_n\) (n=1-9) with F, OH and O, and found that regardless of the thickness, a family of Ti\(_{n+1}C_n(OH)_2\) exhibits ultralow work function. We also found that the changes in the work functions of the functionalized MXenes are linearly correlated with the changes in the surface dipole moments. By considering the three important factors in tuning the surface dipole moments induced by i) the electron charge redistribution at the surface, ii) the surface structure relaxation, and iii) the polarity of attached chemical groups, we have successfully explained the different behavior of the work functions for the F, OH, and O terminated MXenes.

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FIG. 1. The side views of $3 \times 3$ models of (a) the thinnest ($M_2C$) and (b) the thickest ($M_{10}C_9$) MXenes functionalized with OH chemical groups, which were considered in this study. (c) The top views of (a) and (b).
FIG. 2. The work functions of (a) $M_2C$ and $M'_2N$, and (b) $M_{10}C_9$ and $M'_{10}N_9$ ($M = \text{Sc, Ti, Zr, Hf, V, Nb, Ta}$; $M' = \text{Ti, Zr, Hf}$) functionalized with F, OH, and O.
FIG. 3. Work functions of bare and functionalized Ti_{n+1}C_n with F, OH, and O chemical groups.
FIG. 4. The changes in the work functions ($\Delta \phi$) of functionalized M$_{10}$C$_9$ and M'$_{10}$N$_9$ (M= Sc, Ti, Zr, Hf, V, Nb, Ta ; M'= Ti, Zr, Hf) with F, OH, and O as a function of the changes in the surface dipole moments ($\Delta P$). The solid line ($\Delta \phi = -170.01 \Delta P$) indicates the linear fit to the results. The dashed line ($\Delta \phi = -180.95 \Delta P$) indicates the linear relation reported previously for bulk three-dimensional systems [92].
**FIG. 5.** (a) The transferred electron charge density ($\Delta \rho$) between adsorbate (F) and substrate (Hf$_{10}$C$_9$). The vertical dashed lines indicate the encompassing regions of positive ($\ominus$ and $\ominus$) and negative charges ($\ominus$) nearby the surfaces. (b)-(d) the contributions to the total surface dipole moments ($\Delta \mu$): (b) the dipole moment due to the transferred electron charges between surface and the adsorbate ($\Delta \mu$), (c) the dipole moment of the optimized bare surface ($\mu_o$) and the dipole moment of the bare surface without optimization ($\mu_s$) (see the text), and (d) the dipole moment of adsorbate ($\mu_a$). $\Delta \rho$, $\Delta \mu$, $\mu_o$, $\mu_s$, and $\mu_a$ have been calculated as a function of $z$ (along the normal surface) from the center of MXenes to the end of the cell (40 Å), but the results are shown for distances between 0 and 20 Å only. The respective quantities at 40 Å are also indicated in (b)-(d). The circles indicate the positions of the layers and their composed elements.
FIG. 6. The same as Fig. 5 but for Sc\textsubscript{10}C\textsubscript{9}F\textsubscript{2}.
FIG. 7. The same as Fig. 5 but for Sc_{10}C_{9}(OH)_{2}.
FIG. 8. Phonon dispersions of Sc$_2$C(OH)$_2$ and Ti$_2$C(OH)$_2$. 