Impact of Physico-Chemical Parameters on Finfish Eggs Diversity from Muthupettai, South East Coast of India

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Abstract

Physico-chemical parameters can determine the species diversity and abundance in particular area. Environmental parameters such as rainfall, atmospheric temperature, water temperature, salinity, pH and dissolved oxygen were recorded and correlated with the distribution of fish eggs. It is evident from the present study that the water temperature and salinity appear to play a significant role in determining the distribution of fin fish eggs in the study area. The time and intensity of spawning of fishes may perhaps be controlled by the seasonal cycle of the environmental factors. The physico-chemical parameters recorded from various stations presently showed monthly seasonal and annual variations.

Keywords: Fin fish eggs; Physico-chemical; Disturbance; Dispersion; Species survive

Introduction

Predation, food availability [1,2] and environmental parameters have been pointed out as the main factors influencing ichthyoplankton dynamics. Among environmental parameters, temperature, salinity and river flow [3] are considered the most important variables affecting ichthyoplankton communities in estuaries systems. Community level studies supply information on several aspects of species biology, allowing for the determination of spatial and temporal evolution of ichthyoplankton distribution [4], thus establishing preferred spawning grounds and seasons [5]. Moreover, it is also possible to identify factors that influence recruitment variability [6,7]. Estuaries are unstable systems, generally having a limited number of species present. However, they may support high abundance of organisms due to their high productivity, providing important nursery areas where ichthyoplankton encounter suitable conditions for enhanced development [8]. Mangroves possess in varying degrees marine, brackish and fresh water fish components. Their distribution depends upon the geographical biological, ecological and physicochemical characteristics of the water. It is well known that the mangroves are used as nursery and breeding history of fishes. The hydrodynamic conditions in an estuary are quite complex, as they are influenced by river flow, tide, wind and density factors. Further, they are modified due to earth’s rotation, bottom friction and the geometric properties of the estuarine system [9]. Physico-chemical and biological, ecological condition are generally subjects to the eggs and larvae to high rate of mortality in the environment. Hydrological study is a prerequisite to the assessment of the potentialities, distribution of plants and animals and also to understand the realities between its different trophic level and food webs. In Indian estuaries and seas the physicochemical characteristics have been studied by many workers [10-14] in west coast. Major estuarine systems in India are situated on the east coast. Several studies on hydrography of estuarine and coastal waters were carried out from east coast of India [15-18]. Qasim [19] stated that the tropical estuaries are influenced mainly by the monsoonal rain which brings profound changes in the hydrography. The physico-chemical parameters observed in the present study were found to be influenced mostly by the monsoonal rainfall that determines seasons.

Materials and Methods

Monthly water samples were collected from Muthupettai at

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During the year 2010, the water temperature varied between 18.6°C (December) and 29.4°C (January). In station II, the temperature fluctuated from 18.5°C (October) to 29°C (July, September). In station III, it varied from 19°C (October) to 28°C (April, July). At station IV, it ranged between 16.9°C (December) and 28°C (January).

During 2011, the water temperature varied from 16.5°C (November) to 30.1°C (May). At station I, 7.2 (December) and 8.2 (May) in station II, between 7.3 (December) and 8.2 (May) at station III, 7.2 (December) and 8.2 (May) in station IV during the year 2010 (Figure 5).

During 2011, the pH ranged from 7.5 (November) to 8.2 (June) in station I, from 7.7 (November) to 8.2 (May) in station II, from 7.6 (December) to 8.2 (May) in station III, and from 7.2 (December) to 8.2 (May) in station IV during the year 2010 (Figure 5).

The pH values ranged between 7.9 (April) and 8.3 (June) at station I, 7.2 (December) and 8.2 (May) in station II, between 7.3 (December) and 8.2 (May) at station III, 7.2 (December) and 8.2 (May) in station IV during the year 2010 (Figure 5).

During 2011, the pH ranged from 7.5 (November) to 8.2 (June) in station I, from 7.7 (November) to 8.2 (May) in station II, from 7.6 (December) to 8.2 (May) in station III, and from 7.2 (December) to 8.2 (May) in station IV during the year 2010 (Figure 5).
(November) to 8.3 (Jun) in station III and between 7.8 (December) and 8.2 (Jun) in station IV.

The Analysis of Variance (ANOVA) of pH between seasons and between stations showed significant at 0.05% level (Table 3).

Dissolved oxygen

The dissolved oxygen values recorded during the study period (January-2010 to June - 2011) are shown in Figure 6. During (2010), the dissolved oxygen values varied from 5.66 mg/l (February) to 4.11 mg/l (July) in station I, from 5.0 mg/l (February) to 2.8 mg/l (October) at station II, at station III, between 4.5 mg/l (June) and 3.1 mg/l (September). In station IV, the maximum dissolved oxygen value was recorded (5.54 mg/l) during February and the minimum (4.0 mg/l) in November-2010.

During the year 2011, at station I, dissolved oxygen varied between 4.5 mg/l (July) and 3.5 mg/l (December). At station II, dissolved oxygen ranged between 4.5 mg/l (July) and 2.9 mg/l (March). At station III, the dissolved oxygen varied from 4.7 mg/l (July) to 3.1 mg/l (December) and at station IV, it varied ranged from 4.2 mg/l (July) to 3.2 mg/l (October).

The result of ANOVA showed variations in oxygen concentration between stations and between was found to be significant at 0.05 % level (Table 4).

Discussion

The higher rainfall recorded during monsoon months (October to December) and lower values during summer months (April to June). The present trend observed was in agreement with the observations made by earlier workers [21-24].

Variation in atmospheric temperature can influence the physico-chemical characteristics of coastal and estuarine waters to a great extent. The higher values of atmospheric temperature in summer and lower values in monsoon observed in the present study confirmed the established trends along the southeast coast and is in agreement with the observations made by Rajasekar [21], Saravanan [22], Rajaram [23], Sundaramanickam [24], Ganesan [25], Sampath Kumar [26].

The surface water temperature largely depends upon the intensity of solar radiation, evaporation, freshwater influx, cooling and mixing due to currents and tidal flow. Surface water temperature also showed a similar trend of monthly variation as that of air temperature and the high values were reported during the summer and lower values during the monsoon season. The gradual increase in water temperature from monsoon to summer is directly related to atmospheric conduction and radiation. Similar findings were reported by Rajasekar [21], Saravanan [22], Rajaram [23], Sundaramanickam [24], Ganesan [25].

Salinity is considered to be the basic and prime factor among the environmental variables in the marine environment which influences greatly the dynamic situation of the estuarine and coastal waters by the influx of freshwater and the prevailing temperature. It is one of the most fluctuating parameters, typified with wide range of variations in both estuarine and coastal environments. Generally, the changes in the salinity in the brackish water habitats such as estuary, backwater, mangroves and coastal waters are due to the influx of freshwater from river, by land runoff caused by monsoon, or by tidal variations. The maximum salinity was recorded during the summer and the minimum during monsoon months during the study period. Higher values in summer could be attributed to high degree of evaporation with decreased freshwater inflow and land drainage. Drop in salinity during monsoon is related to heavy showers and consecutive floodwater from

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**Table 1:** Two-way ANOVA for differences in surface water temperature between seasons along Muthupettai coastal waters during the study period (January-2010 to December-2011).

| Source of Variation | SS     | df | MS     | F      | P-value | F crit |
|---------------------|--------|----|--------|--------|---------|--------|
| Rows                | 1140.513 | 23 | 49.58754 | 23.47393 | 4.01E-24 | 1.686897 |
| Columns             | 48.04583 | 3  | 16.01528 | 7.58137  | 0.00187  | 2.737492 |
| Error               | 145.7592 | 69 | 2.112452 |         |         |        |
| Total               | 1334.318 | 95 |         |         |         |        |

**Table 2:** Two-way ANOVA for differences in Salinity between seasons along Muthupettai coastal waters during the study period (January-2010 to December-2011).

| Source of Variation | SS     | df | MS     | F      | P-value | F crit |
|---------------------|--------|----|--------|--------|---------|--------|
| Rows                | 2620.99 | 23 | 113.9561 | 13.7652 | 1.29E-17 | 1.686897 |
| Columns             | 10.03125 | 3  | 3.34375  | 0.403906 | 0.750645 | 2.737492 |
| Error               | 571.2187 | 69 | 8.278533 |         |         |        |
| Total               | 3202.24  | 95 |         |         |         |        |

**Table 3:** Two-way ANOVA for differences in pH between seasons and stations along Muthupettai coastal waters during the study period (January-2010 to December-2011).

| Source of Variation | SS     | df | MS     | F      | P-value | F crit |
|---------------------|--------|----|--------|--------|---------|--------|
| Rows                | 9.536562 | 23 | 0.414633 | 11.88923 | 5.83E-16 | 1.686897 |
| Columns             | 0.006146 | 3  | 0.002049 | 0.058742 | 0.981173 | 2.737492 |
| Error               | 2.406354 | 69 | 0.034875 |         |         |        |
| Total               | 11.94906 | 95 |         |         |         |        |

**Table 4:** Two-way ANOVA for differences in DO between seasons and stations along Muthupettai coastal waters during the study period (January-2010 to December-2011).

| Source of Variation | SS     | df | MS     | F      | P-value | F crit |
|---------------------|--------|----|--------|--------|---------|--------|
| Rows                | 13.88135 | 23 | 0.603537 | 2.988639 | 0.00024 | 1.686897 |
| Columns             | 14.73547 | 3  | 4.911822 | 24.32272 | 7.53E-11 | 2.737492 |
| Error               | 13.93412 | 69 | 0.201944 |         |         |        |
| Total               | 42.55094 | 95 |         |         |         |        |
The Vellar estuary [26,27,30,31].

The hydrogen-ion concentration (pH) often changes with time due to temperature-salinity changes and biological activity. The pH remained alkaline throughout the study period in all the stations with maximum values during the summer, which could be attributed to the high salinity of water and also due to the uptake of carbonate by the photosynthetic organisms. The photosynthetic activity may cause high pH, because of bicarbonate degradation by carbonic anhydrase associated with photosynthesis [28].

The hydrogen-ion concentration (pH) often changes with time due to exposure to biological activity and temperature. The low values of pH noticed during monsoon season may perhaps be due to dilution and mixing of coastal waters by rain floods that lead to reduction in salinity and temperature and decomposition of organic matter. The present findings are in agreement with that of Hydrobiological studies in the gradient zone of Muthupettai waters by rain floods that leads to reduction in salinity and temperature. The low dissolved oxygen concentration observed during summer may be attributed to the higher salinity of the water, higher temperature and less inflow of freshwater coupled with biological processes such as consumption of available oxygen by the organisms for respiration and active decomposition of organic matter during summer month [35]. During 2010, the dissolved oxygen concentration was higher compared to 2011, which may perhaps be due to higher rainfall recorded during that period. The trend noticed in the present study is in accordance with the findings of hydrology and heavy metals [23-27, 30].

There is an increasing interest in the recent years to enhance the fishery productivity of a scientific basis. The growth, morphological development and ontogeny of fish eggs and larvae of commercially important fish species have to be studied well to enhance the productivity through the aquaculture practices. Special attention is being paid to biochemical, physiological, biotechnological and hormonal manipulation of gonadal development, sex determination, sex reversal and production of fertilisable gametes as well as to development, growth and brood stock management in relation to eggs and larval quality. Finfish eggs are needed in the aquatic life as they play basic role in the biodiversity of larvae and adult fishes. The landing of fishes which is increase or decrease is depending on the environmental conditions. However, the sampling sites were observed that, manmade and anthropogenic activities can affect the quality of water. There was determining the decrease a living capacity of adult fish and exclude egg and larval fish survival.

Dissolved oxygen is one of the most important parameters, which reflects the physical and biological processes of water. The dissolved oxygen content depends upon the photosynthetic activities, monsoonal floods and the turbulence caused by winds [32]. The dissolved oxygen concentration depends directly on the monsoonal dissolved oxygen and also on the nature and abundance of the planktonic organisms; salinity etc. In the present study, the dissolved oxygen concentration was low during summer and high during monsoon months in all the stations of Muthupettai waters. Hung [33] opined that temperature is a major factor controlling oxygen saturation in water. Solubility of oxygen in water is inversely proportional to temperature [34]. The low dissolved oxygen concentration observed during summer may be attributed to the higher salinity of the water, higher temperature and less inflow of freshwater coupled with biological processes such as consumption of available oxygen by the organisms for respiration and active decomposition of organic matter during summer month [35].

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