Research Article

Chromatography Based Metabolomics and In Silico Screening of Gymnema sylvestre Leaf Extract for Its Antidiabetic Potential

Shabana Parveen,1 Mohd Hafizur Rehman Ansari,2 Rabea Parveen,1 Washim Khan,2 Sayeed Ahmad,2 and Syed Akhtar Husain1

1Human Genetics Laboratory, Department of Bioscience, Jamia Millia Islamia, New Delhi 110025, India
2Bioactive Natural Product Laboratory, School of Pharmaceutical Education and Research, Jamia Hamdard, New Delhi, India

Correspondence should be addressed to Sayeed Ahmad; sahmad.jh@yahoo.co.in

Received 10 June 2018; Revised 3 October 2018; Accepted 6 December 2018; Published 6 January 2019

1. Introduction

Diabetes mellitus (DM) describes a metabolic disorder characterized by a deficiency in insulin production and its action or both [1]. It is thriving distributed in nearly all countries and constantly increases in numbers and implication, as varying quality of life lead to reduced physical activity and increased obesity in populations. That leads to prolonged hyperglycemia with variabilities in most metabolic processes inside the human body [2]. As per global concern World Health Organization (WHO), 347 million people worldwide are suffering from DM, with the estimate that it will be the seventh leading cause of death in 2030. A total of 1.5 million deaths are directly triggered by diabetes in 2012. It was the eighth leading cause of death among both sexes and the fifth leading cause of death in women. When chewed, the fresh leaves of G. sylvestre have the outstanding property of paralyzing the sense of sweet taste substance for few times. The gymnemic acid molecules in terms of atomic arrangements are analogous to that of glucose molecules. These types of molecules fill the receptor location on the taste buds thereby stopping its activation by sugar molecules existing in the food. This, up-to-date study showed that the most general medicinal plants with remarkable antidiabetic importance in terms of their mechanism and modes of action together with the methodology part used for their quality, safety, and efficacy assessment to explore the biological standardization of thousands of traditionally used medicinal plants both in vitro and in vivo studies to establish its phytopharmacological and therapeutic effect.
and in vivo metabolomics approach with chromatographic profiling to assess the claimed activity with the aim of finding potent antidiabetic markers from the natural resources [3]. In Indian systems of medicine, i.e., Ayurveda, the G. sylvestre prominently used in the therapy of dyspepsia, constipation, and hyperglycemia [4] hemorrhoids, jaundice vesicle, renal calculi, asthma, cardiopathy [5] amenorrhea, bronchitis, and leukoderma [6, 7]. The ethanolic extract of G. sylvestre leaves showed the presence of eleven different isomers of gymnemic acids with different molecular weights (gymnemic acid I to gymnemic acid XI). The major phytoconstituents found in G. sylvestre are gymnemic acid (GA), gudmarine, and saponins. Gymnemic acid is a pentacyclic triterpenoid, the main active principle displaying antidiabetic activity [6]. The plant derived extract of G. sylvestre has already reported to have direct insulinothropic activities on β cells and isolated islets of human in vitro [8]. Moreover, antidiabetic potential, G. sylvestre, has the capability of total cholesterol and lower triglyceride in serum and its antiatherosclerotic potential were almost similar to that of a standard lipid-lowering agent clofibrate. Some studies reflected the ability of G. sylvestre to inhibit the formation of advanced glycation end products and sorbitol accumulation [9].

Due to the presence of specific metabolites, it has been used for different therapeutic purposes. Furthermore, these herbal materials have the significant application for various phytopharmacological applications. Several herbal preparations containing the dried leaves of G. sylvestre or its extract are being used for various therapeutic purposes. These plant materials are being used in traditional system of medicine for different disease especially in diabetes. Antidiabetic potential of G. sylvestre leaves has been reported but its metabolomic characterization has not been fully explored. Less scientific data are available on the bioactive metabolites responsible for its antidiabetic activity. In our study, we have qualitatively analyzed the number and category of metabolites present in extract through LC-MS and identified the bioactive metabolites through in silico screening. Further, we have tested that hydroalcoholic extract has been tested for its antidiabetic potential using in vitro and ex vivo approaches. In this context our study provides solid scientific evidence in support of its antidiabetic activity. We have authenticated and extracted the leaves of G. sylvestre.

2. Methodology

2.1. Plant Material and Extract Preparation. The leaves of G. sylvestre obtained from Botanical Garden of Jamia Hamdard, New Delhi, and authenticated as per the standard protocol specified in Ayurvedic Pharmacopoeia. The authenticated plant materials have been deposited in the Bioactive Natural Product Laboratory for future reference with a voucher specimen number JH/SUPER/BNPL/Shabana/2014/GS. The plant sample was washed, shade dried, and coarsely powdered. The powdered drug materials (200 g) of G. sylvestre were defatted with petroleum ether and extracted through soxhlation using 70% (v/v) alcohol for 24 h. The hydroalcoholic extract was concentrated to 1/4th of its volume by rotary vaporization under reduced pressure. The extract was filtered and subjected to basify to isolate gymnemic acid enriched extract using chloroform. The extractive value and % yield of extract were calculated and stored at 4°C for bioactivity and quantitative analysis.

2.2. Total Phenolic and Flavonoid Content. Through the Folin-Ciocalteu method, the total phenolic content in the hydroalcoholic extract of G. sylvestre was determined according to the procedure described in the literature [10]. Different concentrations of gallic acid solutions (as the standard equivalent of phenol) were used for establishing the calibration curve which was further used for the determination of phenol content. All the experiments were carried out in triplicate. The obtained regression equation from the calibration plot was used for the determination of total phenolic content and expressed as mg of gallic acid equivalent per gram of extract. Aluminum chloride (AlCl₃) colorimetric method was used for the determination of total flavonoid content [10]. The total flavonoid content in the hydroalcoholic extract was calculated from a calibration curve of standard (rutin) by using its different dilutions concentrations ranging from 10 to 100 µg/mL. The total flavonoid content was expressed as mg/g of rutin equivalent.

2.3. Determination of Antioxidant Potential. The antioxidant potential of the extract was determined by 1, 1-diphenyl-2-picryl-hydrazyl (DPPH) assay. The stock solution of different concentrations of extract (10, 20, and 40 µg/mL) was mixed with 1.0 mL of a methanolic solution of DPPH (1.0 µg/mL) and incubated for 30 min in dark at room temperature. Then, the absorbance was recorded at 517 nm using a UV-visible spectrophotometer. Trolox was used as standards for comparison.

2.4. In Vitro Carbohydrate Digesting Enzyme Inhibition Assay. For the determination of the α-amylase inhibitory potential of extract, previously developed method was followed [10]. Briefly, accurately weighed 5 mg of enzyme was dissolved in 10 mL of 20 mM phosphate buffer (pH 6.9) at 37°C, while the extract was dissolved in dimethylsulfoxide and diluted in phosphate buffer. Different concentration of extract ranging from 100 to 1000 µg/mL was used for the determination of α-amylase inhibition potential. One mL of diluted extract and 1.0 mL of enzyme solutions (0.5 mg/mL) were mixed together and incubated at room temperature for 30 min. After the completion of incubation, 1.0 mL of 0.5% (w/v) starch solution was added to the mixture and then again kept for 10 min at room temperature. About 2.0 mL of dinitrosalicylic acid was added to the reaction mixture and heated in boiling water for 5 min to stop the previously ongoing reaction. The resulting mixture was cooled and absorbance was measured calorimetrically at 565 nm.

For the determination of α-glucosidase inhibitory potential of extract, previously reported procedure [10] was followed. Extract prepared for α-amylase assay was used for α-glucosidase assay also. While an enzyme solution (1.0 U/mL) was prepared in 10 mM phosphate buffer (pH
6.8). Briefly, the 100 μL of diluted extract and 200 μL of enzyme solution were incubated at 37°C for ten minutes. Then, 100 μL of p-nitrophenyl-α-D-glucopyranoside (PNPG) solution (5.0 mM) in 10 mM phosphate buffer (pH 6.8) were added to start the reaction and the mixture was incubated at 37°C for 30 min. Then the reaction was stopped after the addition of 2.0 mL of 0.1 M sodium carbonate (Na2CO3). Finally, the absorbance was recorded at 405 nm of the yellow colored p-nitrophenol freely released from p-nitrophenyl-α-D-glucopyranoside. Acarbose was used as the positive control and the results were expressed as the inhibition rate (%) of enzymatic activity and figured by the beneath equation:

\[
\text{% Inhibition of enzyme activity} = \left( \frac{\text{Abs}_{\text{control}} - \text{Abs}_{\text{sample}}}{\text{Abs}_{\text{control}}} \right) \times 100 \tag{1}
\]

where \text{Abs}_{\text{control}} is the OD of reaction without extract or standard and \text{Abs}_{\text{sample}} is the OD of the reaction of with extract or standard.

2.5. Ex Vivo Glucose Uptake. The inhibition assay of intestinal glucose uptake was determined in rat hemidiaphragm. About overnight fasted rats were used for this assay. Animals were euthanized by anesthesia and dissected to isolate diaphragm. It was immediately dipped in the ice-cold Krebs-Henseleit buffer which was previously equilibrated with 95% oxygen, 5% carbon dioxide in a cylindrical vessel of organ bath. The extract was added in the same compartment and incubated at room temperature for one hour. Further, glucose solution was added and again incubated for 30 min at room temperature. After the completion of incubation, a sample from the supernatant was collected and the unabsorbed glucose was estimated using glucose estimation kit which was commercially utilized. The amount of glucose uptake by per gram of muscle was measured by the following formula:

\[
\text{Muscle glucose uptake} = \frac{(C_1 - C_2)}{g \text{ of muscle tissue}} \tag{3}
\]

where \(C_1\) and \(C_2\) are the concentrations of glucose before and after incubation, respectively.

2.6. Effect of Extract on Glucose Uptake in Yeast Cells. Commercially utilized baker’s yeast (Saccharomyces cerevisiae) was obtained from Institute of Microbial Technology, Chandigarh, India, and used for this assay. The obtained microbes were subcultured in potato dextrose agar medium and further suspension culture was prepared in potato dextrose broth. The culture medium was centrifuged at 3,000×g for 5 min and the cell pellet was washed in distilled water until the supernatant fluids were clear. About 10% v/v of cell pellet suspension was prepared with the supernatant fluid. One mL of the glucose solution was added with 1 mL of extract and incubated at 37°C for 10 min. The reaction was started by adding 100 μL of yeast suspension to the above mixture. The resulting reaction mixture was vortexed and incubated at 37°C for 60 min. After the completion of the mixture, it was centrifuged and the glucose concentration was measured from the supernatant [10]. Different concentrations of glucose (5, 10, and 25 mM) and extract (250, 500, 750, and 1000 μg) were tested. The percentage increase of glucose in yeast cell was determined using the following equation:

\[
\text{Percentage increase in glucose uptake} = \left( \frac{\text{Abs}_{\text{sample}} - \text{Abs}_{\text{control}}}{\text{Abs}_{\text{control}}} \right) \times 100 \tag{4}
\]

where \text{Abs}_{\text{control}} is the reaction without extract and \text{Abs}_{\text{sample}} is the reaction with extract.

2.7. TLC Fingerprinting of Extract. The hydroalcoholic extract was dissolved in HPLC grade methanol and filtered through 0.25 μM membrane filter. The chromatography analysis was performed on aluminum TLC plates coated with 0.2 μM layers of silica gel 60F254 (Merck Millipore, Germany). Samples were applied with 4.0 mm wide band and 8.3 mm gap between each band by the use of a Linomat V sample applicator (CAMAG, Switzerland). The sample concentration was 10 mg/mL and 5.0 μL sample was applied with a constant sample application rate of 5.0 μL. For the best separation of metabolites, chloroform: methanol (95:05, %v/v) was used as a mobile phase and the plate was developed in a 20 × 10 cm twin-trough glass chamber with linear ascending mode up to 80 mm. Further, the plate was removed from chamber and air dried. The developed plate was scanned at two different wavelengths, i.e., 254 and 366 nm with a TLC scanner III (CAMAG, Switzerland) with slit dimension of 4.0 × 0.30 mm, and the scanning speed was 10 mm/s. The sample application and scanning were operated.
by winCats software. The developed method was validated as per the ICH guidelines for quality control of herbal drugs and botanicals [12, 13]. The peak areas of triplicate samples were used for analyzing the metabolic diversity of hydroalcoholic extract.

2.8. Ultraprecision Liquid Chromatography-Mass Spectrometry Analysis of the Extract. Water’s ACQUITY UPLC™ system (Serial No. #F09 UPB 920M; Model code # UPB; Waters Corp., MA, USA) equipped with a binary solvent delivery system, column manager, an auto sampler, and a tunable MS detector (Serial No # JAA 272; Synapt; Waters, Manchester, UK) was used for UPLC-MS analysis of extract. The extract was chromatographically separated in previously degassed mobile phase consisting of 0.5% v/v formic acid in water (A) and acetonitrile (B) in gradient elution mode (initially 100% A and hold for 5 min; further, decreased to 5% in water (A) and acetonitrile (B) in gradient elution mode respectively. Water’s ACQUITY UPLC™ BEH C18 (100.0 × 2.1 mm × 1.7 μm) column was used and flow rate of mobile phase was 0.4 mL/min. The column manager and sample manager temperature were set to 35 ± 2 °C and 25 ± 2 °C, respectively. About 10 μL of sample was injected with the split mode of 5:1 with the help of autoinjector and the pressure of the system was set to 15000 psi.

The separated metabolites were detected by MS detector on a quadrupole orthogonal acceleration time of flight tandem mass spectrometer (Waters Q-TOF Premier TM). The nebulizer gas and cone gas were set to 500 L/h and 50 L/h, respectively. The source temperature of MS detector was set to 100 °C. The capillary voltage and cone voltage were set to 3.0 kV and 40 kV, respectively. For collision of ion, argon gas was used at a pressure of 5.3 × 10⁻⁵ Torr. The Q-TOF Premier™ was operated in scan mode with resolution over 8500 mass with 1.0 min scan time and 0.02 s interscan delay. Both UPLC and the mass detector were operated by using Mass Lynx V 4.1 software incorporated with the instrument. The separated compounds were identified based on their m/z value through literature survey [10].

2.9. In Silico Screening. To understand the binding mechanisms of active constituents of gymnema leaves, molecular modelling studies were accomplished for deacgymnemenc acid, gymnemic acid, quercetin, and the aglycone moiety gymnemagenin with target proteins by the mopac 6 software package (Stewart Computational Chemistry, Colorado Springs, USA). Different proteins were presumed to interact with targeted molecules (deacgymnemenc acid, gymnemic acid, quercetin, and gymnemagenin). Nine different proteins such as (dipeptidyl peptidase, glucokinase, glutamine fructose-6-phosphate amidotransferase, AMP kinase, GLUT-2, stearoyl-coA desaturase, GLUT-4, sulfonylurea receptor, and mitochondrial Na+/K+ exchanger) were selected for docking analysis. All the docking calculations were achieved on different protein models. In AutoDock tools, solvation parameters, essential hydrogen atoms, and Kollman united atom type charges were added. Autogrid program was employed for generation of affinity (grid) maps of × A grid points and 0.375 Å spacing. Van der Waals and the electrostatic terms were generated by AutoDock parameter set and distance dependent dielectric functions, respectively. Simulations of docking were executed using the Lamarckian genetic algorithm and the Solis and Wets local search method. Initial position, orientation, and torsions of the ligand molecules have been randomly selected. During docking, all rotatable torsions were released. A translational step of 0.2 Å was used, whereas 5 quaternion and torsion steps were utilized in each search. In each docking experiment, two different runs were set and it was terminated after the assessment of maximum 250000 energy was reached. The structure of molecules in mol format was generated in the CDX format using the tool ChemDraw Ultra 7.01 (CambridgeSoft Corporation, Cambridge, USA) and transformed to input ligand format (pdb) for docking by OpenBabel version 2.3.2 Open Babel: An open chemical toolbox (Journal of Cheminformatics 2011, 3:33).

3. Results

The dried leaves of G. sylvestre were defatted through pet ether (60-80 °C). The hydroalcoholic extract (70%) was prepared from soxhlation process. The percentage yield of 70% hydroalcoholic extract was found to be 24.30% w/w. The extract was filtered and basified to isolate the gymnemic acid enriched fraction by successive solvent selection process using chloroform with 1.31% w/w yield. Further, the extract was dried and stored at 4 °C until use.

3.1. Phenolic and Flavonoid Content of Extract. The total phenolic and flavonoid content of the hydroalcoholic extract was determined from the calibration curve of gallic acid ($r^2 = 0.998$) and rutin ($r^2 = 0.989$), respectively. The total phenolic content was found to be 29.36 mg of gallic acid equivalents per gram of extract, while flavonoid content was 18.65 mg of rutin equivalents per gram of extract. This extract is enriched with phenols and especially flavonoids which are mainly responsible for the antioxidant potential of extract. The antioxidant activity was due to the presence of free hydroxyl group present in flavonoids of extract. Free hydroxyl group scavenges the free radicals caused better antioxidant potential. The antioxidant activity of flavonoids, which include flavones, flavanols, and condensed tannins, depends on the presence of free (OH) hydroxyl groups, especially 3-OH, since the present report of antioxidant activity of hydroalcoholic extract suggesting a complete profiling via phytochemical and metabolomics profiling needs to be done to identify the other active phenolic and flavonoid components in the field of drug discovery and development.

3.2. Antioxidant Potential. Due to the simplicity in a biochemical reaction, the free radical scavenging activity of any plant extract is commonly used to determine by DPPH radical. Hydroalcoholic extracts are the rich sources for phenols and flavonoids, which are having redox properties with antioxidant potential. Our phytochemical screening revealed that it has the major abundance of phenolic and flavonoid metabolites. The hydroalcoholic extract had clearly shown the strong antioxidant potential against all free radicals. The
results clearly showed that DPPH scavenging activity was $14.84 \pm 0.12$, $21.14 \pm 0.20$, and $34.36 \pm 0.45$ of mMTR equivalent, at a concentration of 10, 20, and 40 $\mu$g/mL of G. Sylvestre extract, respectively, while that of the control, i.e., ascorbic acid was $41.36$ mMTR when the concentration was $35 \mu$g/mL. The antioxidant potential which was equivalent to DPPH scavenging was increased with respect to the extract concentration and it was increased to a concentration of $40 \mu$g/mL. Beyond the level of extract used, no increment in antioxidant potential was observed. The hydroalcoholic extract enriched with flavonoids which have the potential to scavenge the free radicals associated with different ailments and biological cycle, singlet oxygen, and other oxidizing molecules. Apart from the scavenging the free radicals, flavonoids suppress the production of reactive oxygen species, quenched the trace elements, and upregulate antioxidant defenses which are directly involved in the production of free radical. Similar actions were also reported in extract enriched with phenolic content.

3.3. Carbohydrate Digesting Enzyme Inhibition Potential. The different concentration of $25 \mu$g/mL, $50 \mu$g/mL, and $100 \mu$g/mL of hydroalcoholic extract clearly showed that the % inhibition of $\alpha$-amylase and $\alpha$-glucosidase activity consistently increases with concentration-dependent manner, respectively. The hydroalcoholic extract has clearly shown the remarkable potential of carbohydrate-digesting enzyme at a specific concentration. Carbohydrate digesting enzyme ($\alpha$-glucosidase) inhibitory potential significantly increased with increase in the concentration of hydroalcoholic extract. The carbohydrate enzyme inhibition potential of extract has been shown in Figure 1.

3.4. Yeast Cell Uptake. By facilitated diffusion process in baker’s yeast follows glucose transport process. The process in which glucose was uptaken by skeletal muscle was similar to the process glucose transport in a yeast cell. After specific incubation time period in experimental medium, the amount of the glucose remained in the process of glucose uptake by the yeast cell. It consistently increased in a dose-dependent manner. Figure 2 clearly shows that the percent increases in glucose uptake in yeast cells at different glucose concentrations, i.e., $10 \text{ mmol/L}$, $20 \text{ mmol/L}$, and $30 \text{ mmol/L}$ with respect to concentrations of extract. A concentration-dependent glucose uptake was increased in a yeast cell in the presence of extract. In case of positive control, metformin was used and it also showed the increased glucose uptake in yeast cell. However, hydroalcoholic extract showed greater effectiveness in glucose uptake by yeast cells as compared to positive control. Glucose uptake in yeast cell followed diffusion process which was similar to glucose uptake in skeletal muscle. Thus, these results indicate that hydroalcoholic extract will increase the glucose uptake in skeletal muscle or it will cause an increment in peripheral glucose utilization, while a significant difference in glucose uptake in yeast cell was observed when it was incubated with extract as compared to yeast cell incubated with metformin.

3.5. Ex Vivo Antidiabetic Potential. Generally, researchers have used the isolated diaphragm to check the effect of metabolites on the intestinal glucose inhibition. In our study, we have checked the effect of hydroalcoholic extract of G. sylvestre leaves on intestinal glucose absorption. A marked decrease in glucose absorption in the intestine by extract was recorded as compared to control. A dose-dependent glucose absorption inhibition was recorded (Figure 3). A maximum glucose absorption inhibition (76.25%) has been observed at a concentration of $100 \mu$g/mL of extract. Similar results were obtained in glucose uptake assay in skeletal muscle. The effects of the extract on glucose uptake in isolated rat skeletal muscle are shown in Figure 3. A maximum 28.6% glucose uptake was increased at a dose of $100 \mu$g/mL of extract as compared to control.
3.6. TLC Fingerprinting Analysis. For TLC fingerprinting analysis, hydroalcoholic extract was dissolved in methanol, filtered, and analyzed through 0.25 𝜇M membrane filter. Chromatographic separation was performed by using chloroform: methanol: formic acid, 9:5:0.5, v/v/v, as the mobile phase. The developed plate was scanned at two different wavelengths, 254 nm and 366 nm (Figure 4). The TLC analysis of hydroalcoholic extract clearly showed the separation of total 15 metabolites. Scanning at 254 and 366 nm, a number of compounds analyzed are 12 and 13, respectively (Table 1). However, major components in terms of peaks area in chromatogram were found at Rf 0.01 (98.9%), 0.03 (1.0%), 0.05 (2.0%), 0.07 (2.0%), 0.12 (40.3%), 0.16 (3.0%), 0.23 (4.9%), 0.25 (1.3%), 0.31 (2.1%), 0.35 (1.9%), 0.60 (3.3%), 0.69 (13.7%), 0.79 (14.7%), 0.85 (9.99%) and 0.91 (5.94%) at 254 nm. While at 366 nm, major compounds found at Rf 0.07 (20.23%), 0.12 (22.51%), 0.60 (8.53%), 0.69 (23.62%), and 0.79 (8.79%). One major peak was found at Rf 0.07 (20.23%) while scanning at 366 nm but this peak was not found at 254 nm. Similarly, one peak at Rf 0.91 (5.94%) was found when scanned at 254 nm but found absent at 366 nm.

3.7. UPLC-MS Analysis. All the metabolites present in the hydroalcoholic extract were dissolved in methanol and, based on this assumption, the methanolic solution was analyzed through UPLC-MS for their complete metabolic profiling and identification of diversity of metabolites. Table 2 summarizes all the metabolites characterized in G. sylvestre extract eluted at different retention times, experimental m/z, and tentative name with nature of compounds. A total of 58 most abundant metabolites were analyzed and identified through m/z value and from literature survey. By comparing chromatogram of blank with the chromatogram of extract, a clear and contrast chromatogram was observed based on their retention time (Figure 5). LC-MS fingerprinting of the G. sylvestre extract was done using 0.5% v/v formic acid in water (A) and acetonitrile (B) as mobile phase in gradient elution mode. A total of 58 metabolites were separated and tentatively identified from the database. Decyl gymnemic acid and quercetin were the two major metabolites found in the extract with m/z value of quercetin, conferol, acetylglucosaminic acid, niacin, ascorbic acid, rutin, kaempferol, niacin, D-queritin, and stigmasterol. Mass spectra of major metabolites have been shown in Figure 6.

Major groups of tentatively identified metabolites are alkaloids (25.8%), amino acids (1.7%), coumarins (5.2%), fatty acids (5.2%), flavonoids (25.8%), glycosides (5.2%), lignans (3.5%), lipids (3.4%), nucleosides (5.2%), phenols (5.2%), terpenoids (13.7%), and vitamins (5.2%) (Figure 7). Our investigations revealed that the hydroalcoholic composed of major phenolic and flavonoid metabolites which are responsible for its therapeutic potential. Thus, for the analysis of low and high abundant metabolites with wide polarity range, UPLC-MS method seems to be the best method of analysis.

3.8. In Silico Screening. Gymnemic acid, decyl gymnemic acid, and quercetin were docked with ten different proteins associated with glucose metabolism, transport, and glucose utilization. The docking of tested metabolites with targeted proteins was therefore performed, and corresponding fitness scores were determined. High fitness scored metabolites were subjected to the elucidation of their interaction surface and total intermolecular energy with targeted molecules and proteins separately [14]. Four ligands (deacyl gymnemic acid, gymnemagenin, gymnemic acid, and quercetin) were selected for this study. All these metabolites showed good interaction with glucose transporter and deacylgymnemic acid showed better affinity as compared to other metabolites. Almost all the compounds showed good affinity and the estimated free energy of ligand-protein interaction was found < -5.0 for the receptor proteins in this interaction indicating that affinity of these proteins towards targeted metabolites might be changed after oral administration of extract.
| Metabolites | m/z   | Name                               | Class of compound                  | Reference               |
|------------|-------|------------------------------------|------------------------------------|-------------------------|
| M1         | 350.15| Andrographolide                    | Labdane triterpenoid               | BML80745                |
| M2         | 367.23| Curcumin                           | Flavonoid                          | TY000081                |
| M3         | 457.17| (-)-Epigallocatechin gallate       | Polyphenol 3ds                     | TY000083                |
| M4         | 293.2 | Gingerol                           | Terpenoids                         | CO000211                |
| M5         | 267.14| 3-Hydroxy-3'-methoxyflavone        | Flavonoids                         | BML80385                |
| M6         | 351.23| Ajmalicine                          | Indole alkaloids                   | FIO00001                |
| M7         | 277.2 | Ascorbic acid magnesium phosphate  | Vitamin                            | PubChem CID:10164363    |
| M8         | 1471.4| Cordurol                           | Polyphenol 3ds                     | PubChem CID: 136345     |
| M9         | 365.21| Isopentenyl-adenine-7-glucoside    | Pyridine alkaloids                 | CE000239                |
| M10        | 251.15| (3aR)-(+) -Sclareolide              | Terpenoids                         | BML80075                |
| M11        | 303.07| Dihydroquercetin                   | Flavonoids                         | BML8120                 |
| M12        | 349.22| Strychnine,N-Oxide                 | Labdane diterpenoid                | CO000416                |
| M13        | 275.3 | Eserine                            | Quinidine alkaloid                 | KO008958                |
| M14        | 323.2 | Quinine                            | Coumarin                           | SM884302                |
| M15        | 230.27| 7-Diethylamino-4-methylcoumarin    | Coumarin                           | SM884302                |
| M16        | 335.24| Berberine                          | Isoquinoline alkaloids             | KO008886                |
| M17        | 316.31| Capillarins                        | Sesquiterpenoids                   | TY000038                |
| M18        | 336.24| Lobeline                           | Piperidine alkaloid                | BML81620                |
| M19        | 365.17| Isopentenyl-adenine-7-glucoside    | Pyridine alkaloids                 | CE000239                |
| M20        | 413.29| S,R-Noscapine                      | Benzylisoquinoline alkaloid        | CE000163                |
| M21        | 291.18| Karanjin                           | Steroidal alkaloid                 | BML81520                |
| M22        | 277.24| Linolenic acid                     | Steroidal alkaloid                 | BML81605                |
| M23        | 333.22| Strychnine                         | Labdane diterpenoid                | WA000648                |
| M24        | 302.33| Hesperetin                         | Flavonoids                         | BML81380                |
| M25        | 337.25| 8-Geranyloxy psoralen              | Furanocoumarin                     | BML80640                |
| M26        | 318.32| Myricetin                          | Flavonoids                         | TY000149                |
| M27        | 275.22| Eserine                            | Neolignanoids                      | KO008958                |
| M28        | 301.16| Hematoxylin                        | Neolignanoids                      | BML81375                |
| M29        | 319.25| Coptisine                          | Benzo[c]phenanthridine alkaloids   | TY000086                |
| M30        | 149.05| Methionine                         | Sulphur containing amino acid      | CE000452                |
Table 2: Continued.

| Metabolites | m/z  | Name                                      | Class of compound                  | Reference     |
|-------------|------|-------------------------------------------|------------------------------------|---------------|
| M31         | 382.46 | Dihydrozeatin-9-beta-D-glucoside          | Purine nucleoside                  | PR020117      |
| M32         | 279.25 | Linoleic acid                             | Unsaturated fatty acids            | EQ331601      |
| M33         | 365.29 | Isopentenyl-Adenine-7-glucoside-[d6]      | Pyridine alkaloids                 | CE000594      |
| M34         | 444.43 | Bufotalin (Saponin)                       | Saponin                            | TY000016      |
| M35         | 124.12 | Orcinol                                   | Flavonoids                         | BML81850      |
| M36         | 151.15 | Cathine                                   | Phenylpropanes                      | EQ333501      |
| M37         | 165.14 | D-Quercetiol                              | Flavonoids                         | PubChem CID: 44437 |
| M38         | 177.09 | Ascorbic acid                             | Vitamin                            | PubChem CID: 54670067 |
| M39         | 207.16 | Anthraquinone                             | Flavonoids                         | PubChem CID: 6780 |
| M40         | 251.16 | 7-Hydroxy-3-methylflavone                 | Dihydrochalcone (Phenol class)     | BML80610      |
| M41         | 274.31 | Phloretin                                 | Flavonoids                         | CE000168      |
| M42         | 301.18 | Quercetin                                 | Pyrrolidine alkaloids              | FIO00235      |
| M43         | 335.25 | Seneconine                                | Benzopyran alkaloids               | CE000020      |
| M44         | 351.25 | 4-Methylumbelliferylglucuronide           | Phenylisouquinoline alkaloids      | CE000133      |
| M45         | 353.26 | Chelidonine                               | Flavonoids                         | BML80780      |
| M46         | 353.3  | Asarinin                                  | Benzo furan type lignan            | PubChem CID: 5280794 |
| M47         | 413.28 | Stigmasterol                              | Benzo furan type lignan            | PubChem CID: 5280794 |
| M48         | 429.4  | Ononin                                    | Isoflavonoid                       | PR020043      |
| M49         | 523.39 | 1-Stearoylglycerophosphocholine           | Glycerophospholipids               | MT000126      |
| M50         | 549.44 | Quercetin-3-(6'-malonyl)-glucoside        | Flavonoids                         | PRI01032      |
| M51         | 579.45 | Naringin                                  | Flavonoids                         | CE000186      |
| M52         | 593.3  | Kaempferol-3-O-β-glucopyranosyl-7-O-α-rhamnopyranoside | Glycerophospholipids | PRI01010      |
| M53         | 639.34 | Demethoxycentaureidin 7-O-rutinose        | Flavonoids                         | BML8075S      |
| M54         | 682.4  | Deacyl gymnemic acid II                   | Triterpenoidal saponins            | PubChem CID: 4444284 |
| M55         | 693.52 | Rutin 3’-malonate                         | Flavonoids                         | PubChem CID: 8556617 |
| M56         | 763.56 | Gymnemic acid IV                          | Flavonoids                         | PubChem CID: 4444284 |
| M57         | 835.59 | Triacylglycerol 16:0-16:0-18:0            | Unsaturated fatty acids            | UT000540      |
| M58         | 877.51 | Triacylglycerol 18:2-18:2-18:2            | Unsaturated fatty acids            | UT000521      |
has been observed that gymnemic acid was more potent than deacyl gymnemic acid in terms of binding affinity towards proteins and showed a favorable interaction with amino acid residues at the active site. Docking summary of metabolites docked with proteins associated with diabetes has been shown in Table 3. Some of the major protein interactions are shown in Figure 8. Thus, the present in silico screening study gives an insight of tentatively identified metabolites with a specific protein. The data obtained from in silico screening will help to define/predict the mechanism behind the antidiabetic potential of metabolites present in the extract of G. sylvestre through ligand-receptor interaction.
Figure 6: Mass spectrum of major abundant metabolites analyzed through UPLC-MS.

Figure 7: Categorization of analysed metabolites.
The hydroalcoholic extract from G. sylvestre leaf exhibited potent antioxidant activities and increased the activity of enzymes beneficial in the prevention of diabetes.

From ethnopharmacological relevance point of view, the extract of G. sylvestre has been used for therapeutic purposes and current scientific data are available for its several other therapeutic actions such as antioxidant, anticancer, anti-inflammatory, antidiabetic, hypolipidemic, and hypotensive [7]. Our study clearly demonstrated that the hydroalcoholic extract of G. sylvestre was utilized with a well-defined low dose as mentioned in Ayurved Pharmacopeia for its antidiabetic potential.

We have obtained more yield of hydroalcoholic extract as compared by using general extraction procedure through maceration with water yielding and this can be used in industrial scale also. The total amount of phenolic and flavonoid content in the extracts from G. sylvestre leaf was determined. Gallic acid and rutin were used to express the phenolic and flavonoid content, respectively. The content of phenolic and flavonoid found in natural plants is known to have a number of beneficial health effects associated with natural antioxidants suppressing the LDL cholesterol oxidation [16] and decreasing the risk of heart disorder [17]. The hydroalcoholic extract showed potent antioxidant potential and is expressed as Trolox equivalent. The metabolites present in the extract were able to scavenge the DPPH and we obtained the similar type of activities as ascorbic acid showed. The results obtained from antioxidant assay indicate that hydroalcoholic extract of G. sylvestre leaf contains potent antioxidants potential. It has earlier been proved that plants rich in polyphenolic compounds, such as phenolic acids and flavonoids, possess outstanding antioxidant activities [18].

The results clearly showed that the in vitro α-amylase and α-glucosidase inhibition test displayed that the hydroalcoholic extract of G. sylvestre had a potent inhibitory effect and it showed better activity as compared to acarbose as a standard drug. This in vitro finding suggested that the hydroalcoholic extract of G. sylvestre can be able to significantly reduce the postprandial level by inhibiting the activity of α-amylase and α-glucosidase, which are important enzymes in the digestion of the complex carbohydrates into absorbable monosaccharides in the food. Glucose uptake in yeast cell follows the passive diffusion as similar to glucose absorbable monosaccharides in the food. Glucose uptake in yeast cell has been very first time for G. sylvestre extract. This result was further supported by ex vivo glucose uptake assay. The hydroalcoholic extract was able to increase the glucose uptake in skeletal muscle isolated from rats. In the presence of extract, a significant increase in glucose uptake was observed as compared to normal conditions (Figure 3). On the other hand, we have checked the effect of the hydroalcoholic extract on glucose absorption inhibition in intestine. We found that the extract was able to inhibit the glucose absorption in intestine. From both the ex vivo experiments suggest that the extract is able to increase the peripheral glucose utilization and also decrease the glucose absorption. Thus, it will be more powerful antidiabetic medicine for the patients who are suffering from type 2 diabetes [21]. This ex vivo study provided the biological environment and gives the possible mechanism of the hydroalcoholic extract of G. sylvestre for its antidiabetic potential.

![Figure 8: (a) 3D and (b) 2D interaction of deacylgynemic acid with DPP4.](image)
Table 3: Docking summary of major abundant analyzed metabolites with different proteins associated with diabetes.

| Major metabolites       | Parameters                  | P-1  | P-2  | P-3  | P-4  | P-5  | P-6  | P-7  | P-8  | P-9  |
|-------------------------|-----------------------------|------|------|------|------|------|------|------|------|------|
|                         | Free energy (kcal/mol)      | -6.33| 59.4 | -13.17| -9.22| 27.07| -8.83| -11.08| -5.57| 31.67|
| Deacylgymnemic acid     | Interact Surface            | 599.5| 614.3| 963.6| 859.5| 778.9| 1079.2| 1032.1| 669.8| 698.1|
|                         | Intermolecular Energy (kcal/mol) | -7.94| 59.1| -13.97| -10.36| 26.35| -8.94| -11.77| -6.96| 19.28|
| Gymnemagenin            | Free energy (kcal/mol)      | -8.49| -2.56| 59.4| -10.85| 19.98| -8.91| -10.81| -7.59| -8.39|
|                         | Interact Surface            | 593.8| 646.1| 614.3| 733.8| 481.6| 626.8| 689.5| 558.3| 501.2|
|                         | Intermolecular Energy (kcal/mol) | -8.79| -2.86| 59.1| -11.15| +19.69| -9.21| -11.11| -7.89| -8.69|
| Gymnemic acid           | Free energy (kcal/mol)      | -7.78| -4.29| -9.89| -16.59| -12.49| -9.44| -13.92| -7.97| -3.77|
|                         | Interact Surface            | 582.2| 718.1| 559.8| 824.1| 596.3| 977.7| 819.5| 633.3| 594.3|
|                         | Intermolecular Energy (kcal/mol) | -9.74| -4.35| -9.06| -17.85| -14.13| -11.47| -15.94| -9.6  | -2.61|
| Quercetin               | Free energy (kcal/mol)      | -5.6 | -3.76| -7.46| -7.77| -7.18| -6.08| -8.82| -7.40| -6.28|
|                         | Interact Surface            | 562.5| 602.1| 613.0| 662.4| 630.2| 520.6| 734.0| 644.6| 624.4|
|                         | Intermolecular Energy (kcal/mol) | -5.9 | -4.06| -7.76| -8.07| -7.48| -6.38| -9.12| -7.70| -6.58|

P-1: DPP4; P-2: glutamine–fructose-6-phosphate transaminase; P-3: AMP-activated protein kinase; P-4: GLUT 2; P-5: SCD1; P-6: GLUT 4; P-7: sulfonylurea receptor; P-8: 11-beta hydroxysteroid dehydrogenase type I; P-9: sodium/potassium/calcium exchanger.
Quality control analysis is one of the major concerns for herbal formulation. TLC fingerprinting is usually used to get the metabolite patterns of any extract so that we can identify the extract in future. If any extract will have the same TLC pattern, it must show the same biological activity and we know its TLC fingerprint. In this context, we have developed the TLC method and analyzed our extract. The developed method was reproducible and we have total 13 metabolites present in the hydroalcoholic extract (Figure 4). This TLC fingerprint can be used for its quality control analysis and regulatory bodies to assure its quality and safety [12]. Further, to identify the different metabolites present in the extract, we performed UPLC-MS analysis. UPLC-MS is the most powerful tool for the identification of polar and nonpolar metabolites. In our experiment, we chromatographically separated and tentatively identified based on their mass. Total 58 metabolites were analyzed and identified through UPLC-MS (Table 2). Further, identified metabolites were categorized (Figure 7). HPTLC fingerprinting and LC-MS analysis identify the constituents present in the hydroalcoholic extract which usually polar secondary metabolites such as glycosides, phenols, nucleosides, terpenoids, vitamins, lipids, lignans, fatty acids, saponin, and tannins and some primary metabolites such as glycosides, vitamins, and proteins. Metabolomics is a useful and powerful tool for the chemical and pharmacological standardization of plant extract [22] and it has the potential to make a revolution in research of natural product and to advance the scientifically development of herbal based medicine. Metabolomes of some important medicinal plants are particularly a valuable natural resource for the evidence-based development of new nutraceuticals and phytotherapeuticals. This is the first time we identified the metabolites present in hydroalcoholic extract which were further categorized. This study will be helpful for the researchers who are working in *G. sylvestre*.

In order to predict the mechanism, we have checked the affinity of major metabolites present in the extract with the proteins associated with glucose biosynthesis, metabolism, and utilization. We have selected total four major metabolites present in extract and identified through UPLC-MS. These metabolites were docked with protein and we have analyzed their affinity (Table 3). It has found that gymnemic acid is more potent than deacylgymnemic acid in terms of affinity towards selected protein. It was observed that almost all compounds are with low binding energies and clearly shows that every compound for the enzyme was found with good affinity. *In silico* study of responsible metabolites gives an insight of tentatively identified metabolites with a specific protein. The *in silico* approach can be used to model the interaction and binding energy between a small molecule and protein at the atomic level, which allows us to characterize the behavior of metabolites in the binding site of target proteins as well as elucidate the preliminary mechanism of action of molecules.

In summary, we chromatographically characterized the hydroalcoholic extract of *G. sylvestre* leaves extract and we have tested its antidiabetic potential by using *in vitro, ex vivo*, *in silico*, and metabolomics approaches. The results of the study will be helpful for the development of phytopharmaceuticals which can be used for the management of diabetes.

5. Conclusion

The present study substantiated the hypoglycemic potential of *G. sylvestre* leaves, which has been used since long for the management of diabetes. The hydroalcoholic extract exhibited the hypoglycemic activity by increasing the glucose uptake in skeletal muscle, inhibiting intestinal glucose absorption, and by scavenging redox molecule. *In silico* study predicted the mechanism behind the antidiabetic potential of extract. However, a molecular level study is needed to be performed for better clarification and providing more scientific data. Thus, hydroalcoholic extract enriched with gymnemic acid and deacylgymnemic acid can be explored for the development of phytopharmaceuticals.

Data Availability

The data used to support the findings of this study are available from the corresponding author upon request.

Conflicts of Interest

The authors wish to confirm that there are no known conflicts of interest associated with this manuscript.

Authors’ Contributions

Sayeed Ahmad and Syed Akhtar Husain equally contributed to this paper. The authors are responsible for the content and writing of the paper.

Acknowledgments

The authors would like to acknowledge the University Grants Commission (UGC), New Delhi, for providing Fellowship to Shabana Parveen to carry out the research work.

References

[1] A. N. Singab and F. S. Youssef, "Medicinal plants with potential antidiabetic activity and their assessment," *Medicinal and Aromatic Plants*, vol. 3, no. 1, pp. 1–12, 2014.

[2] J. E. Shaw, R. A. Sicree, and P. Z. Zimmet, "Global estimates of the prevalence of diabetes for 2010 and 2030," *Diabetes Research and Clinical Practice*, vol. 87, no. 1, pp. 4–14, 2010.

[3] I. B. Abdel-Farid, M. G. Sheded, and E. A. Mohamed, "Metabonomic profiling and antioxidant activity of some Acacia species," *Saudi Journal of Biological Sciences*, vol. 21, no. 5, pp. 400–408, 2014.

[4] J. Hess, J. W. Kaderite, and P. Vargas, "The colonization history of Olea europaea L. in Macaronesia based on internal transcribed spacer 1 (ITS-1) sequences, randomly amplified polymorphic DNAs (RAPD), and intersimple sequence repeats (ISSR)," *Molecular Ecology*, vol. 9, no. 7, pp. 857–868, 2000.

[5] D. C. Joshi, P. K. Shrotoria, R. Singh, M. K. Srivastava, and H. S. Chawla, "Assessment of RAPD and ISSR marker systems
Evidence-Based Complementary and Alternative Medicine

for establishing distinctiveness of forage Sorghum (Sorghum bicolor L. Moench) varieties as additional descriptors for plant variety protection,” Indian Journal of Genetics and Plant Breeding, vol. 71, no. 1, pp. 25–36, 2011.

[6] A. K. Verma, S. S. Dhawan, S. Singh, K. A. Bharati, and Jyotsana, “Genetic and chemical profiling of Gymnema sylvestre accessions from central India: its implication for quality control and therapeutic potential of plant,” Pharmacognosy Magazine, vol. 12, supplement 4, pp. S407–S413, 2016.

[7] P. Tiwari, B. N. Mishra, and N. S. Sangwan, “Phytochemical and pharmacological properties of Gymnema sylvestre: an important medicinal plant,” BioMed Research International, vol. 2014, Article ID 830285, 18 pages, 2014.

[8] A. Al-Romaiyan, A. J. King, S. J. Persaud, and P. M. Jones, “A novel extract of Gymnema sylvestre improves glucose tolerance in vivo and stimulates insulin secretion and synthesis in vitro,” Phytotherapy Research, vol. 27, no. 7, pp. 1006–1011, 2013.

[9] L. Kishore and R. Singh, “Protective effect of Gymnema sylvestre L. against advanced glycation end-product, sorbitol accumulation and aldose reductase activity in Homoeopathic Formulation,” Indian Journal of Research in Homoeopathy, vol. 9, no. 4, article 240, 2015.

[10] W. Khan, R. Parveen, K. Chester, S. Parveen, and S. Ahmad, “Hypoglycemic potential of aqueous extract of Moringa oleifera leaf and in vivo GC-MS metabolomics,” Frontiers in Pharmacology, vol. 8, pp. 1–16, 2017.

[11] C. I. Chukwuma and M. S. Islam, “Effects of xylitol on carbohydrate digesting enzymes activity, intestinal glucose absorption and muscle glucose uptake: a multi-mode study,” Food and Function, vol. 6, no. 3, pp. 955–962, 2015.

[12] M. N. Mallick, M. Singh, R. Parveen et al., ”HPTLC analysis of bioactivity guided anticancer enriched fraction of hydroalcoholic extract of Picrorhiza kurroa,” BioMed Research International, vol. 2015, Article ID 513875, 18 pages, 2015.

[13] M. Singh, E. T. Tamboli, Y. T. Kamal, W. Ahmad, S. H. Ansari, and S. Ahmad, “Quality control and in vitro antioxidant potential of Coriandrum sativum Linn.,” Journal of Pharmacy and Bioallied Sciences, vol. 7, no. 4, pp. 280–283, 2015.

[14] R. B. Kumar and M. X. Suresh, “A computational perspective of molecular interactions through virtual screening, pharmacokinetic and dynamic prediction on ribosome toxin A chain and inhibitors of Ricinus communis,” Pharmacognosy Research, vol. 4, no. 1, pp. 2–10, 2012.

[15] D. B. Kitchen, H. Decornez, J. R. Furr, and J. Bajorath, “Docking and scoring in virtual screening for drug discovery: methods and applications,” Nature Reviews Drug Discovery, vol. 3, no. 11, pp. 935–949, 2004.

[16] A. S. Meyer, O.-S. Yi, D. A. Pearson, A. L. Waterhouse, and E. N. Frankel, “Inhibition of human low-density lipoprotein oxidation in relation to composition of phenolic antioxidants in grapes (Vitis vinifera),” Journal of Agricultural and Food Chemistry, vol. 45, no. 5, pp. 1638–1643, 1997.

[17] P. Tiwari, R. S. Sangwan, Asha et al., “Molecular cloning and biochemical characterization of a recombinant sterol 3-O-glucosyltransferase from Gymnema sylvestre R.Br. catalyzing biosynthesis of steryl glucosides,” BioMed Research International, vol. 2014, Article ID 934351, 14 pages, 2014.

[18] R. A. Larson, “The antioxidants of higher plants,” Phytochemistry, vol. 27, no. 4, pp. 969–978, 1988.

[19] V. P. Cirillo, “Mechanism of glucose transport across the yeast cell membrane,” Journal of Bacteriology, vol. 84, pp. 485–491, 1962.

[20] I. C. Selvaraj, “In vitro investigation of antidiabetic potential of selected traditional medicinal plants,” International Journal of Pharmacognosy and Phytochemical Research, vol. 6, no. 4, pp. 856–861, 2014.

[21] G. Jiang and B. B. Zhang, “Glucagon and regulation of glucose metabolism,” The American Journal of Physiology—Endocrinology & Metabolism, vol. 284, no. 4, pp. E671–E678, 2003.

[22] G. Ulrich-Merzenich, H. Zeitler, D. Jobst, D. Panek, H. Vetter, and H. Wagner, “Application of the “Omic” Technologies in phytomedicine,” Journal of Natural Remedies, vol. 7, no. 1, pp. 1–18, 2007.