**Pseudomonas aeruginosa** 4-Amino-4-Deoxychorismate Lyase: Spatial Conservation of an Active Site Tyrosine and Classification of Two Types of Enzyme

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**Abstract**

4-Amino-4-deoxychorismate lyase (PabC) catalyzes the formation of 4-aminobenzoate, and release of pyruvate, during folate biosynthesis. This is an essential activity for the growth of Gram-negative bacteria, including important pathogens such as *Pseudomonas aeruginosa*. A high-resolution (1.75 Å) crystal structure of PabC from *P. aeruginosa* has been determined, and sequence-structure comparisons with orthologous structures are reported. Residues around the pyridoxal 5’-phosphate cofactor are highly conserved adding support to aspects of a mechanism generic for enzymes carrying that cofactor. However, we suggest that PabC can be classified into two groups depending upon whether an active site and structurally conserved tyrosine is provided from the polypeptide that mainly forms an active site or from the partner subunit in the dimeric assembly. We considered that the conserved tyrosine might indicate a direct role in catalysis: that of providing a proton to reduce the olefin moiety of substrate as pyruvate is released. A threonine had previously been suggested to fulfill such a role prior to our observation of the structurally conserved tyrosine. We have been unable to elucidate an experimentally determined structure of PabC in complex with ligands to inform on mechanism and substrate specificity. Therefore we constructed a computational model of the catalytic intermediate docked into the enzyme active site. The model suggests that the conserved tyrosine helps to create a hydrophobic wall on one side of the active site that provides important interactions to bind the catalytic intermediate. However, this residue does not appear to participate in interactions with the C atom that undergoes an sp² to sp³ conversion as pyruvate is produced. The model and our comparisons rather support the hypothesis that an active site threonine hydroxyl contributes a proton used in the reduction of the substrate methylene to pyruvate methyl in the final stage of the mechanism.

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**Introduction**

Aminodeoxychorismate lyase catalyzes the formation of para-aminobenzoate (PABA) from 4-amino-4-deoxychorismate (Figure 1) as part of the biosynthetic route to folate [1][2]. The enzyme, encoded by the **pabC** gene, is labeled as PabC.

In *Escherichia coli* and many other bacteria, PABA is synthesized in two steps starting from chorismate [3]. The first step produces 4-amino-4-deoxychorismate [4] by transferring ammonia, derived from glutamine, to chorismate. This reaction is performed by PABA synthase, a heterodimer of proteins encoded by the **pabA** and **pabB** genes [5]. 4-amino-4-deoxychorismate is then converted to PABA with the loss of pyruvic acid (Figure 1). The pyridoxyl 5’-phosphate (PLP) dependent PabC catalyzes this second step [2][3][6][7]. In the Gram-positive *Bacillus subtilis* a third chemical step is required for PABA biosynthesis. Chorismate is first converted to 2-amino-2-deoxyisochorismate before production of 4-amino-4-deoxychorismate, which is then converted to PABA and pyruvate by PabC [8].

PABA is required for the biosynthesis of folic acid and by extension also of essential metabolites such as thymidylate and methionine. In addition, PABA is required for biosynthesis of antibiotics such as candicidin [9] and certain *Streptomyces* even possess additional copies of **pabC**, located in antibiotic production operons [10].

The inhibition of PabC would reduce PABA levels and so deplete the supply of folic acid available to the microorganism. Provided that the microbe cannot obtain sufficient PABA from the cellular environment or by other means these cells would be expected to cease growth and die. This is what is observed in **pabC** knockouts in *E. coli* [7], *Pseudomonas aeruginosa* [11], *Helicobacter pylori* [12] and *Acinetobacter baylyi* [13]. A few anti-microbial agents exert their effect through inhibition of folate biosynthesis. Sulphonamides for example deplete bacterial intracellular folate levels by inhibiting PABA synthase and dihydropteroate synthetase [14]. Combining novel PabC inhibitors with sulphonamides or with other anti-folates such as the dihydrofolate reductase inhibitor trimethoprim, could produce a synergistic anti-microbial effect and new inhibitors of PabC would allow such a hypothesis to be tested.

X-ray crystal structures of PabC from *E. coli* [15], *Thermus thermophilus* [16] and *Legionella pneumophila* [Protein Data Bank...
from genomic DNA (American Type Culture Collection 47085, tobacco-etch virus (TEV) protease cleavage site. The resulting vector (Novagen) that produces a hexahistidine-tag linked via a vector (Invitrogen) then ligated into a modified pET15b expression site in bold)

Potential of this protein for such early stage drug discovery [23]. Infections and structural studies provide useful data to assess the possibility of new therapies targeting a range of microbial various microbes suggests that inhibition of PabC offers the possibility of new therapies targeting a range of microbial infections and structural studies provide useful data to assess the potential of this protein for such early stage drug discovery [23].

We have a particular interest in pathogens for which current treatments are unsatisfactory, for example Gram-negative bacteria such as P. aeruginosa, that are highly adaptable to their environment and which possess efficient and varied drug-resistance mechanisms. P. aeruginosa is often the cause of life-threatening infections in people with cystic fibrosis, burns victims and immunocompromised individuals [24][25]. We describe the construction of an efficient bacterial recombinant expression system, enzyme purification and crystallization protocols, and report the high-resolution crystal structure of PabC from P. aeruginosa (PaPabC). Comparisons with previously determined PaBc structures are presented, concentrating on active site cofactor interactions. Our structure-based comparisons identified the potential for a tyrosine to participate in catalysis and suggested that a re-evaluation of a published mechanism is warranted.

Materials and Methods

Cloning, protein purification and analytical ultracentrifugation

The P. aeruginosa pabC gene (locus tag: PA2964) was amplified from genomic DNA (American Type Culture Collection 47085, strain PA01) by PCR using the following primers:

PaPabC forward primer

5’-CAT ATG CTG GAC TGG GTC GAC-3’ (NdeI restriction site in bold)

PaPabC reverse primer:

5’-GGA TCC TCA GAA ATC CAG GTC G-3’ (BamHI restriction site in bold)

The PCR product was blunt-end ligated into a TOPO cloning vector (Invitrogen) then ligated into a modified pET15b expression vector (Novagen) that produces a hexahistidine-tag linked via a tobacco-etch virus (TEV) protease cleavage site. The resulting plasmid was heat-shock transformed into E. coli BL-21 (DE-3) GOLD cells (Novagen). Cultures were grown at 37°C in 2 L of Luria-Bertani medium supplemented with 100 µg mL⁻¹ carbenicillin. Gene expression was induced, when the OD₆₀₀ reached 0.6, by addition of 1 mM IPTG and growth continued for a further 16 hours at 18°C. Cells were harvested by centrifugation (3000 x g at 4°C for 30 minutes), resuspended in 50 mM Tris-HCl, 200 mM NaCl, pH 7.5, and lysed by passage through a French pressure cell. Cell lysate was clarified by ultracentrifugation (40000 x g at 4°C for 30 minutes) and the supernatant was passed through a 0.2 µm filter.

The lysate was loaded onto an immobilised metal ion affinity chromatography column (5 mL HisTrap HP, GE Healthcare) with the assistance of an FPLC system (Acta Explorer, GE Healthcare). The hexahistidine-tagged protein was eluted from the column by a linear concentration gradient of imidazole and fractions were analysed by SDS-PAGE. Selected fractions were pooled and dialysed into 100 mM HEPES, 500 mM NaCl pH 7.5. Attempts to cleave the hexa-histidine tag by addition of TEV protease had produced large amounts of precipitate so it was decided to leave the purification tag in place. The protein was further purified by size exclusion gel chromatography (Superdex 200 26/60, GE Healthcare). This column had previously been calibrated with molecular weight standards, blue dextran (>2,000 kDa), thyroglobulin (669 kDa), ferritin (440 kDa), aldolase (158 kDa), conalbumin (75 kDa), ovalbumin (43 kDa), carbonic anhydrase (29.5 kDa), ribonuclease A (13.7 kDa) and aprotinin (6.5 kDa); (GE Healthcare; data not shown). Protein purity was assessed by SDS-PAGE, matrix assisted laser desorption/ionisation – time of flight mass spectrometry (MALDI-TOF MS) and electrospray ionisation mass spectrometry.

The protein concentration was determined spectrophotometrically using a theoretical extinction coefficient of 33,460 M⁻¹ cm⁻¹ (Protparam, [26]).

Crystallization, X-ray data collection, structure solution and refinement

PaPabC was crystallized at 20°C by the hanging drop vapour diffusion method. The enzyme was concentrated to 33 mg mL⁻¹ in 100 mM HEPES, 500 mM NaCl pH 7.5. Stock solutions of PLP (100 mM in DMSO) and PABA (200 mM in deionized water) were prepared and added to the PabC solution to give final concentrations of 0.1 mM PLP and 10 mM PABA. The mixture was left to incubate for 3 hrs then crystallization drops were assembled by mixing 1 µL of the protein:ligand mixture with 1 µL of reservoir containing 10% (w/v) PEG 400, 1.8 M ammonium sulfate and 100 mM MES pH 6.5. Clumps of pale yellow,
orthorhombic blocks grew over one week and a fragment of approximate dimensions 0.3 mm × 0.2 mm × 0.1 mm was used for data collection. Addition of cryoprotectant (25% v/v glycerol) was required to allow crystals to be mounted in a gaseous nitrogen cryostream at −173 °C for diffraction experiments. Crystals were initially characterized using a copper anode X-ray generator (Rigaku 007) and an image plate detector (RAXIS IV ++ ). The crystal was stored in liquid nitrogen for data collection at the European Synchrotron Radiation Facility (ESRF, Grenoble, France).

Diffraction data were collected on beamline ID29 at ESRF using a Quantum4 CCD detector. It proved beneficial to use the automated crystal annealing facility established on the station. Data were indexed and integrated using MOSFLM [27], and scaled with SCALA [28]. Crystallographic statistics are presented in Table 1. The crystal was stored in liquid nitrogen for data collection at the European Synchrotron Radiation Facility (ESRF, Grenoble, France).

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with a 90% or greater similarity to each other were grouped and a representative sequence identified. This reduced the list to 197 entries. Sequences consisting of more than 400 amino acids were removed and this further reduced the list to 181 entries. These were aligned using MUSCLE [37] and then inspected following pairwise alignment with PaPabC. A conservative group of 129 putative PabC sequences was then identified following the removal of sequences that were fragmentary, labeled as other proteins or which displayed a sequence identity of less than or equal to 20%.

Molecular modeling of the catalytic intermediate

A model of the substrate, 4-amino 4-deoxychorismate, was prepared using PRODRG [38] and placed into the active site using the molecular graphics program COOT [32] with the position of the bound pyridoxyl 5'-phosphate as a guide to allow formation of the catalytic intermediate. The active site was prepared for docking of this ligand using ICM pro (MolSoft L.L.C; http://www.molsoft.com/index.html) with the centre of the ligand-binding site defined by a cavity that contained the residues within 5 Å of the 4-amino 4-deoxychorismate. The top ten docking poses, as scored by ICM Pro software gave similar orientations in the active site and one was selected for discussion and display.

Results and Discussion

Overall structure

Size exclusion gel filtration, carried out during protein purification, indicated a single species of approximate mass 50 kDa that is commensurate with a homodimeric species. In the crystal structure two polypeptide chains, labelled A and chain B together constitute the asymmetric unit in PaPabC (Figure 3), and the crystals have an estimated bulk solvent content of 45%. The polypeptide chains form a dimer (Figure 3) and, as described below, such oligomerization is required to create the enzyme active site. The presence of a PaPabC dimer is consistent with previous studies since all PabC structures solved to date are dimeric or in the case of T. thermophilus PabC, a dimer-of-dimers is observed [16].

The PaPabC subunits display a high degree of non-crystallographic symmetry. The rmsd following a least-squares fit of 270 Cα positions is 0.4 Å. It is therefore only necessary to detail one subunit and one active site.

The PaPabC subunit is constructed from two domains (Figure 4). The N-terminal domain (domain I) is smaller than the C-terminal domain (domain II). Domain I comprises residues 1–107, which
contains three α-helices (α1–3) and a four-stranded anti-parallel β-sheet in order β1–β4–β3–β2. Domain II comprises residues 112–271 that form five α-helices (α4–6), one short two-stranded parallel β-sheet (β5–β6) and one short two-stranded anti-parallel β-sheet (β7–β8). Residues 108–111 link the two domains. The PLP cofactor is covalently bound to Lys140 that is in domain II. The active site is formed in a cleft formed between the two domains (Figure 3).

Sixty residues from each subunit, provided by both domains, contribute to dimer formation. These residues participate in around 30 hydrogen-bonding associations and eight salt bridge interactions to stabilize the quaternary structure. Most of the residues that contribute to the dimer interface are found in two patches; residues Ala10 to Gly25 centred on α1 and residues Arg127 to Ala152 centered on α5.

The approximate dimensions of the dimer, which constitutes the asymmetric unit, are 80 Å × 40 Å × 40 Å and this displays a solvent accessible surface area of approximately 21,000 Å². The contact area between the two subunits occupies approximately 2,050 Å², about 15% of the surface of a subunit. These values are similar to those observed for other PabC proteins, for example the interface of the E. coli protein (PDB code 1e0t) covers an area of 2,130 Å².

The structures of PabC from three organisms (E. coli, L. pneumophila and T. thermophilus) display a high degree of structural conservation with PabC. Least-squares fitting of the Cz positions of single subunits results in RMSD values of between 1.27 and 1.95 Å, with a coverage in the range of 81 to 92% of the structures. The sequence identity shared with PabC, falls in the range 26% to 35%.

The active site and mechanistic considerations

The active site of PabC is positioned between the two domains of the monomer close to the dimer interface (Figure 3). Here, the PLP cofactor is covalently linked to Lys140, bound and oriented by interactions with key residues, a selection of which are depicted in Figure 2. The PLP phosphate accepts hydrogen bonds donated by a cluster of main chain amide groups from Val200, Met201 and Ser237, and the side chain of Arg46. The side chain of Arg46 is itself positioned by interactions with the side chain of Glu28, which in turn accepts a hydrogen bond from His43 ND1. Two well ordered water molecules mediate hydrogen bonding networks linking the phosphate group to Asn178 OD1 and Arg202 N on one side, the main chain O and N groups of Thr29 on the other. The cofactor ring is sandwiched between Val197 and the main chain of a tripeptide segment comprising Val175-Phe176-Ser177. A hydrogen bond between Asn236 ND2 and the carbonyl of Phe176 in conjunction with hydrogen bonds formed between the amide groups and the side chain of Glu161 helps to determine the conformation of the tripeptide that forms a tight turn, between β9 and β10, that covers part of the cofactor.

The PLP methyl group makes van der Waals interactions with Gln147 CG and is positioned 3.2 Å distant from the carbonyl oxygen of Val115. The geometry of this latter contact is compatible with the presence of a C≡H–O hydrogen bond [39]. The side chain of Gln147 is held in place by a hydrogen bond with Arg144 NE. The arginine side chain is itself positioned by hydrogen bonds formed with the carbonyl groups of Leu139 and His141. There are also three water molecules (data not shown), which form a complex network of hydrogen bonds linking the side chain of Arg144 to other sections of the enzyme. In addition the positive dipole from γ6, the closest residue on this helix is Gly199, provides an attractive force to interact with the PLP phosphate. The side chain of Glu173 displays a degree of conformational freedom with two rotamers being observed. One rotamer places the side chain to accept a hydrogen bond donated from PLP N1 (Figure 2). The alternative rotamer places the carboxylate at a distance of 3.7 Å from N1 (data not shown). Tyr92 from the partner subunit forms a hydrogen bond to PLP O3. The cofactor is also likely to possess an intra-molecular hydrogen bond involving O3 and NZ of Lys140. The side chain of Phe27, Val197 and aliphatic components of Arg46 and Arg144 surround the side chain of Lys140.

No structures of PabC-ternary complexes have yet been published to inform on aspects of substrate recognition and enzyme activity. Given the importance of such complexes we judged it important to get such data and attempted to determine structures of PabC in complex with a number of compounds, for example, the sample used for structure determination was crystallized in the presence of 10 mM PABA. However, we have been unable to identify electron density compatible with binding of any such ligand in the substrate-binding pocket. We judged therefore that the position of PLP and Lys140 NZ, together with the conservation of amino acid sequence and structures give an indication of where substrate binds and which residues are important for substrate recognition and catalytic function. Consideration of an amino acid sequence alignment and structural overlay of PabC with three other structures in the PDB proved particularly informative. To this we added molecular modelling and docking of the covalent intermediate formed during catalysis.

The contributions of 24 residues, within about 5 Å of PLP, to binding the cofactor and organising the active site are described above and summarized in Figure 5. Seven of these residues are strictly conserved in PabC of P. aeruginosa, E. coli, L. pneumophila and T. thermophilus (Phe27, Thr29, His43, Arg46, Gln147, Lys140 and Arg202 in PabC). With one exception these residues contribute to the active site using the chemical properties of their side chains. The exception is Gly199 where any alteration on Cz would result in a steric clash with the phosphate group of PLP. Another seven residues contribute to the active site primarily through the hydrogen bonding capacity of the main chain Val175, Phe176, Ser177, Leu139, His141, Val200 and Met201. Phe176 is not conserved in any of the three other sequences but the remainder are conserved in at least one other orthologue. Ser237 interacts with the PLP phosphate using main chain and side chain groups. This residue is only different in the E. coli sequence where an alanine is observed and an additional water molecule binds nearby helping to satisfy hydrogen-bonding capacity (data not shown). A similar observation is made when considering PabC Asn236. The residue is conserved in EcPabC and LpPabC but changed to glycine in TpPabC where a water molecule is observed to replace the side chain (data not shown). Asn178 is strictly conserved in EcPabC and LpPabC but changed to serine in TpPabC. The side chain here simply helps to form one side of the cofactor-binding site. Val197 is conserved in EcPabC and conservatively substituted as leucine in the other two sequences. The residue at this position in the active site forms van der Waals interactions with the cofactor. In PabC Arg144 NE donates hydrogen bonds to Glu147 OE1 and the carbonyl of His141; interactions that help to create one side of the cofactor-binding site and to fix the position of Val197. These residues and the interactions they form appear to be conserved in EcPabC but not in LpPabC or TpPabC. Sequence and structure alignments (not shown) indicate that PabC Arg144 is replaced by tyrosine in both LpPabC and TpPabC and PabC Glu147 replaced by an alanine in LpPabC and a tyrosine in TpPabC. This is an intriguing observation to be discussed below.

There are three glutamates amongst the 24 residues discussed above, Glu28, Glu161 and Glu173. Glu28 helps to position Arg16 to bind the cofactor phosphate. Both residues are strictly
conserved in \( \text{Lp} \text{PabC} \). A conservative substitution of the glutamate to threonine occurs in both \( \text{Ec} \text{Pabc} \) and \( \text{Tt} \text{PabC} \) and the hydrogen bonding interactions to support the role of an arginine binding the cofactor are preserved. Glu161 helps to configure a tripeptide segment to cover part of the cofactor and this feature is conserved since a glutamate is present in the other structures or, as in \( \text{Lp} \text{PabC} \) an aspartate occupies this position. The carboxylate side chain of Glu173 can interact directly with the cofactor by accepting a hydrogen bond donated from N1. In a similar fashion to what is observed for Glu161, such interactions occur in the orthologues with the residue either strictly conserved, or as in \( \text{Tt} \text{PabC} \) replaced by aspartate.

A catalytic mechanism, based on previous studies of PLP-dependent enzymes, was proposed for PabC prior to any structural data being available [10]. Following structure determination of \( \text{Ec} \text{Pabc} \), a model of substrate binding was prepared to inform an assessment of the mechanism [15]. The contribution of PLP was considered the same, that is, following covalent linkage the enzyme-cofactor adduct undergoes a nucleophilic attack by the substrate amino group to produce a reactive external aldime. This species then undergoes a \( \alpha \)-proton elimination, at C4, to the \( \varepsilon \)-amino group of a lysine to yield a reaction intermediate. In \( \text{Ec} \text{Pabc} \) Lys159 forms the covalent link to the cofactor and likely responsible for \( \alpha \)-proton abstraction. Nakai et al., [15] further

**Figure 5.** Structure-based alignment of \( \text{Pa} \text{PabC} \) and \( \text{Lp} \text{PabC} \). Residues encased in black are strictly conserved. Red stars indicate the tyrosine residues at Position I (\( \text{Pa} \text{PabC} \) Y92) and Position II (\( \text{Lp} \text{PabC} \) Y144). Triangles mark residues that are discussed in the text. Red triangles identify Lys140 and residues that interact directly with the Lys140-PLP adduct; yellow triangles mark residues that contribute to the organization of the active site or that participate in solvent mediated interactions between the protein and the cofactor.

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**Figure 6.** Structural conservation of tyrosine in the PabC active site. PLP is shown in the same manner as Figure 2. Left panel. A structural overlay of \( \text{Pa} \text{PabC} \) (green C atoms) and \( \text{Ec} \text{Pabc} \) (blue C atoms). Here, Tyr92 and Tyr109, which represent position I, are contributed from the partner subunit. Centre panel. \( \text{Lp} \text{PabC} \). Right panel. \( \text{Tt} \text{PabC} \). Residue Thr27 displays two rotamers. In the \( \text{Lp} \text{PabC} \) and \( \text{Tt} \text{PabC} \) structures, Tyr144 and Tyr130 respectively represent position II and belong to the same subunit that forms the PLP-binding site.

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suggested that Thr28 of EcPabC supplies a proton to reduce the olefin moiety of the substrate as pyruvate is released. To recover the Schiff-base form of the enzyme, PABA is released and PLP reacts with Lys159.

Three residues described as contributing to the mechanism of action in EcPabC, Thr28, Lys97 and Lys159 [15], are strictly conserved in PaPabC as Thr29, Lys80 and Lys140. They are also conserved in LpPabC and TpPabC with the conservative change of Lys80 to arginine in the thermophilic orthologue (Figure 5). Lys140 of PaPabC is suitably placed to support α-proton abstraction and although Thr29 may engage in the proposed proton shuffling with both the olefin moiety of the substrate and Lys80 we noted an alternative residue that might fulfill this role. This residue may have been overlooked because Nakai et al., [15] had access to only a single structure of PabC and alignments based on sequences alone would not identify a tyrosine that is conserved in three-dimensions.

In EcPabC and PaPabC tyrosine OH groups (Tyr109 and Tyr92 respectively), from the partner subunit, donate a hydrogen bond to PLP O3 (Figures 2, 6); an interaction that influences the electronic structure of the cofactor [15]. The sequence alignment of PabC orthologues suggests a degree of variation in the identity of the amino acid that occupies the position of this tyrosine (Figure 5). The structural overlays are striking since they clearly identify a conserved tyrosine in relation to the cofactor (Figure 6). In EcPabC and PaPabC, Tyr109 and Tyr92 respectively, are contributed from the partner subunit. We call this position I. However, in LpPabC and TpPabC, Tyr144 and Tyr130 respectively, are contributed from the same subunit, or position II.

We isolated 129 putative PabC sequences to investigate the distribution of tyrosine at these two positions seeking to address the hypothesis that if tyrosine is present at position I then it is likely absent from position II and vice versa. This hypothesis holds true for 125 out of the 129 sequences (data not shown). Two sequences, which share less than 30% sequence identity, have a tyrosine at both positions. Analyses of the sequences suggests that PabC can be classified into two groups depending on whether this active site tyrosine is provided by the same subunit which primarily forms the active site or from a partner subunit.

Furthermore the high degree of conservation of an active site tyrosine in three-dimensional structures of PabC suggests a functional role. One possibility is that this residue could act as a proton donor for the methylene to methyl reduction that completes the reaction. Such a reduction would have little influence on the cleavage of the C-O bond of the substrate since the driving force for that reaction would be production of a conjugated aromatic six-carbon ring system. Our model of the catalytic intermediate (Figure 7) was specifically created to investigate the potential role of this active site tyrosine and to re-evaluate the reduction process that completes the reaction.

The model suggests that Tyr112 donates a hydrogen bond to the C1 carboxylate whilst, with contributions from the partner subunit, the main chain amide of Tyr92 and hydroxyl group of Tyr22 donate hydrogen bonds to the other carboxylate of the intermediate (Figure 7). Tyr92 contributes to the positioning and electronic structure of the cofactor by virtue of the hydrogen bond involving the hydroxyl group with the PLP carbonyl (Figure 2). The bulk of the tyrosine side chain serves to create a hydrophobic wall on one side of the active site. Although placed to interact with the catalytic intermediate this residue does not appear to be involved in donating a proton to convert methylene to methyl as part of the catalytic cycle (Figure 7). Rather, Thr29, as first postulated by Nakai et al., [15] appears to fulfill such a role with the hydroxyl group placed about 3.5 Å from the olefin (Figure 7). Lys80 NZ is around 4.3 Å distant from the sp² hybridized C and it is possible that an alternative rotamer could bring this functional group into play with respect to either providing a proton itself or in activating the Thr29 hydroxyl for proton donation. We note that Thr29 is strictly conserved in 123 out of the 129 PabC sequences and conservatively substituted by serine in another five sequences. The single sequence in which we could not identify conservation in terms of a threonine or serine corresponding to Thr29 is the putative PabC from Pantoea vagans. This is a poorly conserved sequence sharing approximately 30% identity.

To further investigate the structure-activity relationship of PabC, and to assess the potential of this enzyme as a target for structure-based drug discovery will require the powerful combination of steady-state kinetic analysis with site-directed mutagenesis studies, and to carry out screening of chemical libraries. It will be imperative to extend from our work and to elucidate a means whereby structural data on enzyme-ligand complexes can be obtained.
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Author Contributions

Conceived and designed the experiments: PEFO TCE WNH. Performed the experiments: PEFO TCE PKF SMS. Analyzed the data: PEFO TCE PKF WNH. Contributed reagents/materials/analysis tools: SMS. Wrote the paper: PEFO TCE WNH.