Identification of the Constituents of Percutaneous Absorption from *Duhaldea nervosa* Based on UHPLC-Q-Exactive Orbitrap MS and Microdialysis Technique

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*Duhaldea nervosa* (*D. nervosa*) has been used for treatment of bone fracture by external use. Thus, the percutaneous absorption was crucial to the effect of *D. nervosa*, especially the constituents of percutaneous absorption. However, the constituents in vivo were never investigated to date. In this study, an efficient method was developed for the identification of constituents of percutaneous absorption using UHPLC-Q-Exactive Orbitrap MS and microdialysis technique. A total of 20 constituents including 15 chlorogenic acid analogues, 3 amino acids, and 2 organic acids were unambiguously or tentatively identified based on high-resolution mass data including MS and MS², chromatography retention time, and bibliography data. To the best of our knowledge, this is the first study to report the constituents of percutaneous absorption from *D. nervosa*, which will be very helpful for understanding the bioactive compounds and quality control.

1. Introduction

*Duhaldea nervosa* (Wallich ex Candolle) A. Anderberg (*D. nervosa*), which belongs to the plant family Compositae, known as Maoxiucai or Xiaoheiyao in China, is a perennial plant widely distributed in the southwestern region of China and Southeast Asia [1]. Traditionally, it has been used as folk medicine in dispelling wind-chill, alleviating pain, promoting the circulation in meridian and collateral for treating migraine, rheumatism, and traumatic injury, especially in accelerating the healing of a fracture by external use or oral administration [2, 3]. Previous studies showed that the main constituents of this plant are steroids, terpenes, flavones, and chlorogenic acid analogues, which possess a variety of biological activities including anti-inflammatory activity [4–6].

Considering the usage of *D. nervosa*, which has been used as an external drug for the treatment of bone fracture, the percutaneous absorption was crucial to the effect of *D. nervosa*, especially the constituents of percutaneous absorption [7–9]. As far as we know, the constituent in vivo has not been investigated; therefore, it is worthwhile to identify the constituents of percutaneous absorption after *D. nervosa* for external use.

In recent decades, microdialysis has become a very powerful sampling technique that enables monitoring of the small molecules in vivo due to its excellent versatility [10, 11]. However, it is limited by the method of analyzing the resulting dialysate. LC-MS as a new technique was used to analyze and identify the constituent in extract botanical and biological sample including dialysate. Among all exiting platforms, ultra-high performance liquid chromatography (UHPLC) coupled with high resolution mass spectrometry (HRMS), including UHPLC-Q-TOF MS, UHPLC LTQ-Orbitrap MS, and UHPLC Q-Exactive Orbitrap MS, is the most powerful technique for the detection and identification
of constituents, as UHPLC can provide a fast and effective separation while HRMS can provide accurate mass measurement and fragment ion, which will be very much beneficial for structure elucidation [12–14].

The present study was designed to detect and identify the main constituents of percutaneous absorption from *D. nervosa* by UHPLC-Q-Excutive Orbitrap MS and microdialysis. Finally, a total of 20 constituents including 15 chlorogenic acid analogues, 3 amino acids, and 2 organic acids were detected and identified based on high-resolution mass data including MS and MS², chromatography retention time, and bibliography data. To the best of our knowledge, this is the first study to report the constituents of percutaneous absorption from *D. nervosa*, which will be very helpful for understanding the bioactive compounds and quality control.

2. Material and Methods

2.1. Chemicals and Materials. Acetonitrile and formic acid of LC-MS grade and methanol of LC grade were obtained from Aladdin Industrial Corporation. The ultrapure water used throughout the experiment was purified by a Milli-Q water purification system (Millipore, Milford, MA, United States). Other reagents of analytical grade were obtained from Aladdin Industrial Corporation.

The reference standards of 3-cafeoylquinic acid (3-CQA, neochlorogenic acid, X-014-170309), 4-cafeoylquinic acid (4-CQA, cryptochlorogenic acid, Y-067-180425), 5-caveoylquinic acid (5-CQA, CGA, L-007-171216), 3,5-dicaffeoylquinic acid (3,5-DiCQA, isochlorogenic acid A, Y-068-170903), 3,4-dicaffeoylquinic acid (3,4-DiCQA, isochlorogenic acid B, Y-069-180105), and 4,5-dicaffeoylquinic acid (4,5-DiCQA, isochlorogenic acid C, Y-070-170515) were purchased from ChengDu Herbpurify Co., Ltd. (ChengDu, China). The purities of all reference standards were no less than 98% based on HPLC-UV analysis.

2.2. Sample Preparation. The root of *D. nervosa* was extracted by reflux with about eightfold 50% ethanol at 70°C for two hours. Then, the filtrate extracts were concentrated under reduced pressure to yield a black residue. Finally, the residue was redissolved in 50% ethanol to give a sample with a concentration of 0.5 g/mL.

2.3. Animal Experiments. All in vivo microdialysis were performed with CMA 402 syringe pump and MAB 85 refrigerated fraction collector (CMA, Microdialysis AB, Sweden). Four male SD rats (weighing 150–200 g, Hunan SJA Laboratory Animal Company, China) were used in *in vivo* study. All procedures were performed under the conditions of National Act on the Use of Experimental Animals. Anesthesia was induced by an intraperitoneal injection of 1.2 mL/100 g 20% urethane before each experiment. The leg of the rat was shaved carefully with razor without breaking the cuticle. After shaving, the rats were placed on an animal heat insulator. Body temperature was kept at 36–38°C. A CMA 20 Elite microdialysis probe (4 mm, polyarylethersulfone membrane, MWCO of 20 kDa) was then inserted into the dermis after fixing the introducer needle, parallel to the skin on the leg. Probe was perfused with normal saline before the insertion. The inlet and outlet were sealed to keep air from entering the probes.

The probes were perfused with normal saline at a flow rate of 2.0 μL/min. A 60 min blank dialysate sample was collected for release of the insertion microtrauma prior to the application of *D. nervosa*. Then, a dosage of 5 g/mL of the material was applied to an area of 2.0 × 3.0 cm² and covered with gauze and bandages. Microdialysate samples were obtained every 60 min up to 10 h, which was combined into one sample for each rat. During sampling, the microdialysis vials were cooled to 4°C; afterward, they were stored at −80°C until analysis.

2.4. Sample Pretreatment. All the microdialysates were pretreated by a solid-phase extraction (SPE) method. A SPE column (WondaSep C18, 200 mg/3 mL) was activated and equilibrated with 6 mL of methanol and 6 mL of water containing 0.5% formic acid, successively. A total of 1 mL microdialysate was loaded on the column. Then, the column was washed by 3 mL of water containing 0.5% formic acid and 3 mL of methanol, respectively. The methanol elute was collected and concentrated under N₂ at room temperature to gain residue, which was redissolved in 100 μL of acetonitrile-water (1:1, v/v) and centrifuged at 12000 rpm for 30 min at 4°C. Finally, an aliquot of 5 μL supernatant was injected into the UHPLC-Q-Excutive Orbitrap MS.

2.5. Instruments and Conditions. An Ultimate 3000 focused (Dionex, Sunnyvale, CA, USA) system equipped with an online vacuum degasser, a quaternary pump, and an auto sampler was used for UHPLC analysis. Separation was performed on an HYPERSIL GOLD C18 column (100 × 2.1 mm, 1.9 μm) at 35°C. The mobile phase consisted of water containing 0.1% formic acid (A) and acetonitrile (B) using a gradient elution at a flow rate of 0.3 mL/min. The flow gradient was applied: 5%–10% B at 0–2 min; 10%–20% B at 2–5 min; 20%–25% B at 5–10 min; 25%–55% B at 10–12 min; 55%–80% B at 12–15 min; 80%–5% B at 15–16 min; and 5% B at 16–20 min.

All EI-MS² analyses were carried out on a Q-Exactive Focus Orbitrap MS (Thermo Electron, Bremen, Germany) coupled with a heated electrospray ionization source (Thermo Electron, Bremen, Germany) in the negative mode. The tune operating parameters were as follows: the rate of sheath gas flow and auxiliary gas flow was 30 and 10 (arbitrary unit), respectively; spray voltage, 3.0 kV; the temperature of capillary and auxiliary gas heater was 320°C and 350°C, respectively; high-resolution MS² was operated at full scan with a mass range of m/z 100–1200 at a resolution of 35000 and MS² at a resolution of 17500 triggered by data-dependent MS² scanning; nitrogen served as collision gas; and the energy was set as normalized collision energy 30%.
and after application of version 3 using the metabolomics workflow templates to detect compounds. The mass tolerance of the maximum peak intensity was set as 10000; the maximum element was shown in Figure 1.

### 2.6. Data Processing and Analysis.
The Xcalibur software version 4, (Thermo Fisher Scientific, San Jose, CA, USA) was used to acquire the raw data including the full-scan MS and MS² data, which were processed by the Compound Discover version 3 using the metabolomics workflow templates to detect the differential components between the microdialysates before and after application of D. nervosa. The detailed parameters of metabolomics workflow template were as follows: The minimum peak intensity was set as 1000; the maximum element counts were C30 H60 O20 S4 N10 Cl4; the mass tolerance of MS and MS² was within 5 and 10 ppm, respectively; and differential analysis was selected for postprocessing.

### 3. Results and Discussion

#### 3.1. Identification of Percutaneous Absorption Constituents of D. nervosa.
A total of 20 constituents were detected and identified based on UHPLC-Q-Exactive Orbitrap MS and microdialysis technique. The retention time and mass spectrometric data of those constituents are listed in Table 1. The high-resolution extracted ion chromatography of those compounds is shown in Figure 1.

#### 3.2. Identification of Chlorogentic Acid Analogs.
Compounds 3, 6, 7, 17, 18, and 20 with the pseudomolecular ion [M-H]⁻ of m/z 353.08783 (0.07 ppm, C₁₆H₁₇O₈), 353.08771 (~0.27 ppm, C₁₆H₁₇O₈), 353.08780 (~0.02 ppm, C₁₆H₁₇O₈), 515.11981 (0.60 ppm, C₂₅H₂₃O₁₂), and 515.11938 (~0.23 ppm, C₂₅H₂₃O₁₂), respectively, were accurately identified as tran-3-CQA, tran-5-CQA, tran-4-CQA, 3,4-DiCQA, 3,5-DiCQA, and 4,5-DiCQA by comparing their chromatography retention times, accurate mass measurement, and fragment pattern with those data of reference compounds.

Compounds 9, 11, and 19 were eluted at 5.37, 5.69, and 8.16 min, respectively, with the deprotonation ion [M-H]⁻ of m/z 353.08783 (0.07 ppm, C₁₆H₁₇O₈), 353.08771 (~0.27 ppm, C₁₆H₁₇O₈), 353.08780 (~0.02 ppm, C₁₆H₁₇O₈), 515.11981 (0.60 ppm, C₂₅H₂₃O₁₂), and 515.11938 (~0.23 ppm, C₂₅H₂₃O₁₂), respectively, which were confirmed by the MS² spectrum. Finally, those were tentatively characterized as cis-5-CQA, 1,3-DiCQA, and 1,5-DiCQA according to the literature data [15, 16]. Compounds 5, 10, and 15 possessing a [M-H]⁻ ion of m/z 337.09323 (1.01 ppm, C₁₆H₁₇O₈), 337.09302 (0.38 ppm, C₁₆H₁₇O₈), and 337.09286 (~0.09 ppm, C₁₆H₁₇O₈), respectively.
were eluted at 4.12, 5.59, and 6.31 min, respectively, suggesting that they were p-coumaroylquinic acid (pCoQA). Considering the base peak at m/z 163.0389, 191.0552, and 173.0445 of the precursor ion at m/z 337 [17, 18], they were tentatively presumed to be 3-pCoQA, 5-pCoQA, and 4-pCoQA, respectively.

Compounds 8, 14, and 16 eluted at 4.70, 6.13, and 6.64 min, respectively. All of them showed a pseudomolecular ion [M-H]− of m/z 367.10306 (−1.08 ppm, C17H19O9), 367.10355 (0.26 ppm, C17H19O9), and 367.10294 (−1.40 ppm, C17H19O9), respectively. The base peak at 193.0497, 173.0445, and 191.0547 in MS2 spectrum were used to discriminate isomers of feruloylquinic acid (FQA) [17, 18]; therefore, they were tentatively inferred to be 3-FQA, 4-FQA, and 5-FQA, respectively.

3.3. Identification of Amino Acid. Compounds 4, 12, and 13 were detected at 3.45, 5.70, and 6.07 min with the [M-H]− ion at m/z 203.08181 (−3.55 ppm, C11H11O2N2), 172.09697 (−5.05 ppm, C8H14O3N), and 172.09691 (−5.85 ppm, C8H14O3N), respectively. The MS2 spectrum was used to elucidate the structure by searching various MS/MS databases including Massbank, METLIN, and KEGG [19]. Therefore, they were temporarily determined to be tryptophan, N-acetyl-leucine, and N-acetyl-alloisoleucine, respectively.

3.4. Identification of Organic Acid. Compounds 1 and 2 with the same pseudomolecular ion [M-H]− of m/z 133.014 (C4H5O5) were detected at 0.96 and 1.11 min, respectively. The fragment ion at 115.0026 (−9.4 ppm, C3H6O3), 71.0131 (−10.6 ppm, C4H6O2) and 89.0235 (−10.3 ppm, C4H6O2) was detected in MS2 spectrum by natural loss of H2O, CH2O3, and CO2, suggesting the presence of carboxyl and hydroxyl. According to the literature [20], they were temporarily identified as malic acid isomers.

4. Conclusion

The constituents of percutaneous absorption following the external use of D. nervosa were investigated using UHPLC-Q-Exactive Orbitrap MS and microdialysis technique. Finally, a total of 20 constituents including 15 chlorogenic acid analogues, 3 amino acids, and 2 organic acids were detected and identified based on high-resolution mass data including MS and MS2, chromatography retention time, and bibliography data. To our best knowledge, this is the first study to investigate the constituents of percutaneous absorption from D. nervosa, which will be very beneficial for understanding the bioactive compounds and quality control.

Data Availability

The data used to support the finding of this study are available from the corresponding author upon request.

Disclosure

This work was presented as an oral presentation at Consortium for Globalization of Chinese Medicine (CGCM2019) in Shanghai.

Conflicts of Interest

The authors have declared no conflicts of interest.
Authors’ Contributions
Lianghong Liu and Lian Zhu contributed equally to this work.

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