Thermal decomposition of fullerene nanowhiskers protected by amorphous carbon mask

Hongxuan Guo1,2, Chengxiang Wang3, Kun’ichi Miyazawa4, Hongxin Wang5, Hideki Masuda5 & Daisuke Fujita5

Fullerene nanostructures are well known for their unique morphology, physical and mechanical properties. The thermal stability of fullerene nanostructures, such as their sublimation at high temperature is also very important for studying their structures and applications. In this work, we observed fullerene nanowhiskers (FNWs) in situ with scanning helium ion microscopy (HIM) at elevated temperatures. The FNWs exhibited different stabilities with different thermal histories during the observation. The pristine FNWs were decomposed at the temperatures higher than 300 °C in a vacuum environment. Other FNWs were protected from decomposition with an amorphous carbon (aC) film deposited on the surface. Based on high spatial resolution, aC film with periodic structure was deposited by helium ion beam induced deposition (IBID) on the surface of FNWs. Annealed at the high temperature, the fullerene molecules were selectively sublimated from the FNWs. The periodic structure was formed on the surface of FNWs and observed by HIM. Monte Carlo simulation and Raman characterization proved that the morphology of the FNWs was changed by helium IBID at high temperature. This work provides a new method of fabricating artificial structure on the surface of FNWs with periodic aC film as a mask.

In the last century, four types of allotropes of carbon, three-dimensional (diamond and graphite), two-dimensional (graphene), one-dimensional (nanotubes), and zero-dimensional (fullerenes) have been extensively studied. A zero-dimensional fullerene is a small spherical cluster created with 60 or more carbon atoms such as C60, C70, and so on. From low dimension to high dimension, the fullerene molecules form various macro- and nano-crystalline structures via van der Waals bonds, such as fullerene nanotubes (FNTs) and fullerene nanowhiskers (FNWs). One dimensional fullerene nanostructures, especially FNTs and FNWs, are attractive for research since they can be used as FETs, templates for nanostructures, and others. Due to the low van der Waals bonds energy, FNWs are not stable at high temperature or in high energy electron irradiation. As a crystalline structure made of fullerene with size of 1 nm, fullerene nano structure can be manipulated by a top-down nano fabrication in high spatial resolution. However, to the knowledge of the author, artificial structure made on the surface of FNWs by a top down method have not been well investigated.

Different from other top-down nano fabrication methods, electron beam induced deposition (EBID) and ion beam induced deposition (IBID) are fast and resistless deposition techniques which can overcome the fundamental resolution limits of conventional lithography techniques. Various materials can be deposited by using a scanning electron microscope (SEM) via EBID or a focused ion beam (FIB) system via IBID. For example, Pt, W and other materials have been deposited by using a gallium ion source FIB system with a gas injection system (GIS). Scanning helium ion microscopy (HIM) involves using an advanced scanning microscope with the same mechanism but with a higher resolution and greater focal depth than those with SEM and FIB systems. Different materials, such as Pt, PtC, and W were deposited with C6H5Pt, (CH3)3Pt(CpCH3), and WF6.
as precursors respectively, by using a HIM system with GIS\textsuperscript{23–25}. Without GIS, amorphous carbon (aC) can also be deposited by EBID in electron microscope with hydrocarbon as a precursor\textsuperscript{26,27}. In this process, the hydrocarbon absorbed on the surface of the samples decomposes into carbon atoms and is deposited again as aC film when the electron or ion beams scan the surface of the samples.

In this paper, we investigated the thermal decomposition of the FNW with an in situ heating sample holder setting in the HIM. During the scanning by helium ion beam, ion beam induced carbon deposited on the surface of FNW and formed a mask. A periodic structure based on the thermal decomposition of FNWs protected by the mask at high temperatures was observed in situ by the HIM.

Results

Structure of fullerene nano whiskers. The as-grown FNWs synthesized by using a toluene solution of C\textsubscript{60} and isopropyl alcohol (IPA) with an liquid-liquid interfacial precipitation (LLIP) method had a hexagonal structure with a lattice constant of \(a = 2.405 \text{ nm}\) and \(b = 1.001 \text{ nm}\) measured by X-ray diffraction (XRD) and high resolution transmission electron microscopy fast Fourier transform (HRTEM-FFT) pattern. After the solvent evaporation, C\textsubscript{60} FNWs formed with an FCC structure with a cell dimension \(a = 1.423 \text{ nm}\), as indicated by the XRD characterization\textsuperscript{27}. The FNWs annealed at 310 °C for 2 h were also checked by XRD as shown in Fig. 1(a). The diffraction result was indexed by an FCC system. This result is consistent with the results obtained by Minato and Miyazawa\textsuperscript{27}.

The FNWs annealed at different temperatures or irradiated by 30 keV He\textsuperscript{+} ions were characterized by Raman spectroscopy, as shown in Fig. 1(b). Peaks around 272 cm\textsuperscript{−1}, 496 cm\textsuperscript{−1}, and 1470 cm\textsuperscript{−1} can be assigned to H\textsubscript{2} squashing mode, \(\Delta_A\) breathing mode, and \(\Delta_A\) tangential stretching mode of fullerene molecules, respectively\textsuperscript{28,29}. Figure 1(c) shows the details of Raman peaks around 1470 cm\textsuperscript{−1} of FNWs annealed at 310 °C. The red lines in the inset are Raman spectra of as-prepared FNWs. A high intensity peak at 1470 cm\textsuperscript{−1} and a shoulder peak at 1460 cm\textsuperscript{−1} indicate that parts of the sample were polymerized\textsuperscript{28,30}. After annealed at 310 °C and 380 °C, the peak at 1470 cm\textsuperscript{−1} disappeared. Peaks at 1460 cm\textsuperscript{−1} show that all the fullerene molecules in FNWs were polymerized. Since all the samples were characterized by Raman with the same power laser, laser induced polymerization is not considered for contribution\textsuperscript{30}. The polymerization can be attributed to thermal induced polymerization or sublimation of the pristine fullerene molecules\textsuperscript{9,10,28}. After annealed at 490 °C, the peak at 1460 cm\textsuperscript{−1} shifted to 1468 cm\textsuperscript{−1} again. This result is due to the thermal decomposition of the polymerized FNWs and will be discussed in this paper. Similar to the samples annealed at 310 °C and 380 °C, He\textsuperscript{+} irradiated FNW also shows a single peak at 1460 cm\textsuperscript{−1}. The polymerization is attributed to high energy charged particle induce polymerization of C\textsubscript{60} molecules. This is also observed in TEM observation of C\textsubscript{60} films and FNWs\textsuperscript{50,51}.

Thermal decomposition of fullerene nano whiskers. Figure 2(a,b) show a special FNW with two different kinds of surface morphology. Two different areas, \(\Omega\) and \(\Omega'\), are clearly separated by a boundary between them. The sample was checked on the surface of the in situ heating sample holder for two rounds. In the first round, the FNW was scanned at a temperature lower than 300 °C. The aC was deposited on the scan area as indicated by the \(\Omega\) in Fig. 2(a,b), respectively. After that, the FNW was heated to 490 °C. The sublimation of FNW area covered by the deposited aC film was found to be suppressed by the protection effect of the over-layer film. Then, the images shown in Fig. 2(a,b) were obtained with the second round of scan performed at 490 °C.

Figure 2(c–e) show three different typical FNWs with different processes. The FNW in Fig. 2(c) was scanned several rounds from RT to high temperature. In the last round, the image was taken as shown in Fig. 2(c). The fullerene molecules were protected from sublimation by deposited aC films from the surface to the center of the whiskers. The FNW in Fig. 2(d) was also scanned twice. The sample was scanned at 380 °C first and then at 490 °C. During heating, the inner C\textsubscript{60} of the FNW was sublimated by the high temperature and then a fullerene shell generated. The FNW in Fig. 2(e) was observe at 490 °C once and then taken an image. Without protection from deposited aC, the FNW in Fig. 2(e) was totally destroyed by sublimation. Only the broken C\textsubscript{60} residues were left on the substrate.

Discussion

Hollow structures of FNWs similar to Fig. 2(d) have been observed and published by several groups including our previous works. Xu and Fujita et al. found that the FNWs are covered with C\textsubscript{60} oxide layers in an ambient atmosphere by using scanning Auger microscopy and conductive atomic force microscopy\textsuperscript{5}. The different stabilities of the outer C\textsubscript{60} oxide layer and the inner molecular crystal provide a method of creating a shell structure by sublimating the inner C\textsubscript{60} crystal. A C\textsubscript{60} shell structure, called “cocoon”, was first prepared by Haluška et al. by heating C\textsubscript{60} crystal in a vacuum environment\textsuperscript{3}. The shell structure of C\textsubscript{60} was also observed in FNWs by using high resolution transmission electron microscopy (HRTEM) after the FNWs were annealed in a vacuum at 600 °C. The C\textsubscript{60} crystal was totally sublimated at 600 °C according to the HRTEM images and selected-area electron diffraction pattern\textsuperscript{10}. Only the aC shell, or “cocoon” was obtained after heating. The decomposition of FNWs was observed in situ by non-contact atomic force microscopy (NC-AFM) by Fujita et al. The temperature-dependent morphology of a FNW indicates that the FNW was starting to be decomposed and shrunk at \(\sim 300 \text{ °C}\).

Based on the thermal stability of FNWs, it is possible to make nano structure on the surface of FNWs with high spatial resolution surface patterning by HIM. The spatial resolution of ion-beam-induced aC deposition is determined by beam energy, deposition time, scattering of secondary electrons (SEs) in the substrate and as-deposited nanopillars, and so on\textsuperscript{25}. Besides the high energy primary beam, the SEs emitted from the substrate, with energies of less than 50 eV, play an important role in the EBID and IBID, because the cross section for the electron impact dissociation of the hydrocarbon molecules peaks at low energies\textsuperscript{15}. The spatial resolution of SE scattering is defined by the spot size of the helium ion beam on the sample surface and the trajectory of the helium ions scattered in the sample. In this work, the de Broglie wave length of helium ions with an energy of 30 keV is
0.083 pm, as calculated from $\lambda = \frac{h}{\sqrt{(2mQ)}}$, where $h$ is Planck’s constant, $m$ is the mass of helium ions, and $Q$ is the energy of helium ions. The helium ions penetrated the substrate with weak scattering and experienced an electronic loss event. The helium beam was weakly scattered because the ions were lighter than the sample nuclei and much heavier than electrons. This makes the interaction radius between the helium beam and surface of the sample small and the SE yields high\(^3\). The resolution of the SE images obtained by using HIM was measured by scanning the edge of highly oriented pyrolytic graphite (HOPG) as 0.35 nm.

The periodic structure shown in Fig. 2(d) was made by a sequential scans at 380 °C and 490 °C. At first, the FNW shown in Fig. 3(a) was scanned by a helium ion beam with current of 0.5 pA and dwell time of 12 μs, respectively. Every lines in the image was scanned for 32 times. The size of the image is 1024 × 1024 pixels. Therefore, the interval between every lines is about 12 nm. After that, parts of the FNW shown in Fig. 3(c) was scanned twice to show the details of the FNW morphology. Finally, the sample was kept annealing at 490 °C and scanned to obtain the SE images shown in Fig. 3(b,d,e).

During scanning by helium ion beam, amorphous carbon atoms are deposited on the surface by the beam and secondary electrons via IBID. The deposited aC lines protected the $C_{60}$ molecules from the FNW decomposition.

**Figure 1.** FNWs observed by X-ray diffraction and Raman spectroscopy. (a) is the X-ray diffraction results of FNWs annealed at 315 °C for 3 hours. (b) is the Raman spectrum of pristine FNWs and FNWs annealed at 310 °C, 380 °C, and 490 °C. (c) is the Raman peak of $A_g$ mode of FNWs.
It makes periodic structure on the surface of FNW, as found in the SE images obtained at 490 °C. The periodicity is about 11 nm obtained by the line profile and FFT pattern shown in Fig. 4(b,c), respectively. As shown in Fig. 3(b,c), a part of the FNW was scanned twice at 380 °C with field view of 1 μm. The line intervals in Fig. 3(c) is about 1 nm. Since it was scanned twice with a bit shift, the aC lines are overlapped. As a result, aC film covered on the surface of FNW homogeneously and protected the FNW from decomposition. Figure 3(d,e) show the morphology of FNW decomposed at 490 °C. We marked 7 indicative patterns in Fig. 3(c) with different color lines. The indicative patterns were marked in (d) and (e) to show the overlapped areas in Fig. 3(c–d,e-2 and e-2) are the overlapped areas in (c) and (d), (e). They show smooth surface since aC film is homogeneous on the surface of FNW. (d-1), (e-1), and (e-3) are out of the overlapped area as shown by the indicative patterns. These areas were scanned once to obtain Fig. 3(a). The helium ion beam induced deposited aC lines with intervals about 12 nm masked the C60 fullerenes from decomposition. Then, periodic structure was formed in these areas as show in Figs 3(d-1,e-1,e-3) and 2(d).

High energy particles irradiation in the surface and bulk of the FNWs induces the polymerization of C60 molecules, as we characterized by Raman spectroscopy shown in Fig. 1 10,31. The polymerization was also observed in thermal treated FNWs. However, after annealed at a higher temperature as 490 °C, the polymerized C60 molecules decomposed. So, the beam induced polymerization can not form a stable structure on the surface of FNWs. With a Monte Carlo method, we simulated the scattering of He ions in the FNWs with an energy of 30 keV as shown in Fig. 4(d) 33. The typical diameters of FNWs are various from 300 nm to 1 μm as measured by SEM. From the simulation results, we can see that the penetration depth of He ions in the FNW is about 300 nm. Most of the scattering evens are in a wide range more than 200 nm. In this case, the C60 molecules in FNWs are polymerized simultaneously. Therefore, the FNW can not be selectively irradiated by the He ion beam to form a periodic structure.

Based on the Monte Carlo simulation, we can see that only the surface modification, such as helium ion induced carbon deposition, can form a periodical structure on the surface of FNWs. Since the SEs are important to the EBID and IBID, the resolution of the aC deposition is a bit larger than the SE image resolution15. Based on some special techniques, sub 10 nm patterns have been obtained by EBID17,34–36. Typical resolution of electron beam induced deposition with a 2 nm primary beam is about 15–20 nm measured by experiments and Monte Carlo simulation35. Based on its higher secondary electron imaging resolution and surface sensitivity, HIM is advanced
in high spacial resolution surface patterning. In Winston's work, helium ion directly writing lithography with resolution of 20 nm and 10 nm has been introduced.

**Conclusion**

In this work, we studied the FNWs at ambient and high temperatures with HIM. It was found that the FNWs can be sublimated at a high temperature when they are exposed to the vacuum environment. The deposition of aC on the surface of FNWs protected them from decomposition at a temperatures as high as 490 °C in our *in situ*
observations. With Monte Carlo simulation and Raman characterization, we found that helium ion beam induced aC film deposition can change the morphology of the FNWs. Based on the high special resolution, an ultra-thin aC film deposited on the surface of the FNWs by helium IBID of HIM. The fullerene molecules was selectively protected by the aC film from sublimation and form an artificial periodic structure on the surface of annealed FNWs. This experiment confirmed the high spacial resolution of ion beam induced aC deposition with HIM.

Methods

Fullerene nano whiskers preparation. The FNWs were synthesized by using an LLIP method. A toluene solution saturated with C60 was prepared and poured into a glass bottle. Then, isopropyl alcohol was gently added into the bottle to form a liquid-liquid interface between the isopropyl alcohol and the C60-saturated toluene. The bottle was kept at 15 °C. Then, the FNWs were nucleated and grown at the liquid-liquid interface in the glass. After synthesis, the FNWs were transferred onto the Si sheet of the in situ heating sample holder from the alcohol solution. After that, they were observed at room (RT) and high temperatures by the HIM.

Fullerene nano whiskers structure characterization. X-ray diffraction. The FNWs were annealed in vacuum with a pressure better than 1.2 × 10^{-8} Torr at 315 °C for 3 h. After that, they were measured by Rigaku RINT powder diffractometer with monochromatized Cu Kα radiation (λ = 0.15405 nm).

Raman spectroscopy. The FNWs were annealed in vacuum with pressure better than 1.2 × 10^{-8} Torr at 310 °C, 380 °C, and 490 °C for 30 min. Then, the samples including pristine FNWs were characterized by confocal Raman microscope (Renishaw) with a laser with wavelength of 532 nm. The laser power was 45 mW as calibrated by a power meter. The exposure time was 10 s. The FNWs were also irradiated by helium ion beam with energy of 30 keV, beam current of 5.0 pA, scan area of 1.4 × 1.4 mm², for 5 min and checked by the Raman spectroscopy. Same parameters were applied for Raman characterization.
Scanning helium ion microscope and in situ heating sample holder design. The structure of HIM is outlined in Fig. 5(c). A ultra sharp tip was placed in an ultra-high vacuum chamber with only three atoms (trimer) at the apex, called a source, as shown in Fig. 5(a,b). Three helium ion beams were extracted from the trimer and accelerated by acceleration voltage that could be adjusted from 8 keV to 40 keV. Small apertures with diameters of 20 μm, 10 μm, or 5 μm were applied to select one helium ion beam for scanning on the surface of the specimen. An Everhart-Thornley detector was set up in the sample chamber to detect the SEs emitted from the surface of the specimen induced by the inelastic scattering of helium ions and the specimen.

The in situ heating sample holder was designed according to the schematic in Fig. 5(e). A silicon (Si) sheet with dimensions of 20 mm × 2 mm × 0.2 mm was fixed onto a copper sample holder by using tantalum (Ta) sheets. The distance between the Ta sheets was 18 mm. The Si sheet was insulated from the copper sample holder by using four ceramics isolating rings with a height of 2 mm. After being baked at temperatures above 300 °C for 72 h, the holder was placed in the sample chamber of the HIM. Two shielded cables connected the Si sheet with a DC power supply to heat the sample. The temperature at the center of the Si sheet was calibrated by the power of the DC current as plotted in Fig. 5(d).

References
1. Kroto, H. W., Heath, J. R., O’Brien, S. C., Curl, R. F. & Smalley, R. E. C60: Buckminster Fullerene. Nature (London, UK) 318, 162 (1985).
2. Babu, S. S., Möhwald, H. & Nakanishi, T. Recent progress in morphology control of supramolecular fullerene assemblies and its applications. Chem. Soc. Rev. 39, 4021 (2010).
3. Haluška, M., Kuzmany, H., Vybornov, M., Rogl, P. & Fejdi, P. A double-temperature-gradient technique for the growth of single-crystal fullerenes from the vapor phase. Appl. Phys. A-Mater. Sci. Process. 56, 161 (1993).
4. Ogawa, K. et al. Electrical properties of field-effect transistors based on C60 nanowhiskers. Appl. Phys. Lett. 88, 112109 (2006).
5. Xu, M., Pathak, Y., Fujita, D., Ringor, C. & Miyazawa, K. Covered conduction of individual C60 nanowhiskers. Nanotechnology 19, 075712 (2008).
6. Minato, J. & Miyazawa, K. C60 fullerene tubes as removable templates. J. Mater. Res. 21, 529 (2006).
7. Tiwari, R. N., Ishihara, M., Tiwari, J. N. & Yoshimura, M. Synthesis of graphene film from fullerene rods. Chem. Commun 48, 3003 (2012).
8. Wang, Y., Holden, J., xin Bi, X. & Eklund, P. Thermal decomposition of polymeric C60. Chem. Phys. Lett. 217, 413 (1994).
9. Miyazawa, K., Minato, J., Fujino, M. & Suga, T. Structural investigation of heat-treated fullerene nanotubes and nanowhiskers. Diamond Relat. Mater. 15, 1143 (2006).
10. Miyazawa, K. et al. Structure and electrical properties of heat-treated fullerene nanowhiskers as potential energy device materials. J. Eur. Ceram. Soc. 26, 429 (2006).
11. Nakaya, M., Aono, M. & Nakayama, T. Molecular-scale size tuning of covalently bound assembly of C60 molecules. ACS Nano 5, 7830 (2011).
12. Gaevski, M. E. et al. Two-dimensional photonic crystal fabrication using fullerene films. Nanotechnology 11, 270 (2000).
13. Chen, X., Palmer, R. E. & Robinson, A. P. G. A high resolution water soluble fullerene molecular resist for electron beam lithography. Nanotechnology 19, 275308 (2008).
14. Gibbons, P. F., Robinson, A. P. G., Palmer, R. E., Manickam, M. & Preece, J. A. Ultrathin fullerene films as high-resolution molecular resists for low-voltage electron beam lithography. Small 2, 1003 (2006).
15. Silvis-Cividjian, N., Hagen, C. W., Kruit, P., v.d. Stam, M. A. J. & Groen, H. B. Direct fabrication of nanowires in an electron microscope. Appl. Phys. Lett. 82, 3514 (2003).
16. Baglin, J. Ion beam nanoscale fabrication and lithography—a review. Appl. Surf. Sci. 258, 4103 (2012).
17. van Dorp, W. F., van Someren, B., Hagen, C. W. & Kruit, P. Approaching the resolution limit of nanometer-scale electron beam-induced deposition. Nano Lett. 5, 1303 (2005).
18. MoberlyChan, W. J., Adams, D. P., Aziz, M. J., Hobler, G. & Schenkel, T. Fundamentals of focused ion beam nanostructural processing: Below, at, and above the surface. MRS Bulletin 32, 424 (2007).
19. Morgan, J., Notte, J., Hill, R. & Ward, B. An introduction to the helium ion microscope. Microscopy Today 14, 24 (2006).
20. Bell, D. C. Contrast mechanisms and image formation in helium ion microscopy. Microsc. Microanal. 15, 147 (2009).
21. Guo, H., Gao, J., Ishida, N., Xu, M. & Fujita, D. Characterization of two-dimensional hexagonal boron nitride using scanning electron and scanning helium ion microscopy. Appl. Phys. Lett. 104, 031607 (2014).
22. Guo, H., Itoh, H., Wang, C., Zhang, H. & Fujita, D. Focal depth measurement of scanning helium ion microscope. Appl. Phys. Lett. 105, 023105 (2014).
23. Sanford, C. A. et al. Beam induced deposition of platinum using a helium ion microscope. J. Vac. Sci. Technol. B 27, 2660 (2009).
24. Chen, P. et al. Nanopillar growth by focused helium ion beam-induced deposition. Nanotechnology 21, 455302 (2010).
25. Daryl A Smith, D. C. J. & Rack, P. D. Monte Carlo simulation of focused helium ion beam induced deposition. Nanotechnology 21, 175302 (2010).
26. Broers, A. N., Molzen, W. W., Cuomo, J. J. & Wittels, N. D. Electron-beam fabrication of 80 Å metal structures. Appl. Phys. Lett. 29, 596 (1976).
27. Minato, J. & Miyazawa, K. Solved structure of C_{60} nanowhiskers. Carbon 43, 2837 (2005).
28. Sathish, M. & Miyazawa, K. Synthesis and characterization of fullerene nanowhiskers by liquid-liquid interfacial precipitation: Influence of C_{60} solubility. Molecules 17, 3858 (2012).
29. Bethune, D. S. et al. Vibrational raman and infrared spectra of chromatographically separated C_{60} and C_{70} fullerene clusters. Chem. Phys. Lett. 179, 181 (1991).
30. Kato, R. & Miyazawa, K. Raman laser polymerization of C_{60} nanowhiskers. Journal of Nanotechnology 2012, 101243 (2012).
31. Zhao, Y. B., Poirier, D. M., Pechman, R. J. & Weaver, J. H. Electron stimulated polymerization of solid C_{60}. Appl. Phys. Lett. 64, 577 (1994).
32. Cohen-Tanugi, D. & Yao, N. Superior imaging resolution in scanning helium-ion microscopy: A look at beam-sample interaction. J. Appl. Phys. 104, 063504 (2008).
33. Ziegler, J. F., Biersack, J. P. & Ziegler, M. D. SRIM: The Stopping and Range of Ions in Matter (SRIM Co, Chester, Maryland, 2012).
34. van Dorp, W. F. Sub-10 nm writing: focused electron beam induced deposition in perspective. Appl. Phys. A 117, 1615 (2014).
35. Fujita, J. et al. Carbon nanopillar laterally grown with electron beam-induced chemical vapor deposition. J. Vac. Sci. Technol. B 21, 2990 (2003).
36. Synchugov, I., Nakayama, Y. & Mitsuki, K. Sub-10 nm crystalline silicon nanostructures by electron beam induced deposition lithography. Nanotechnology 21, 285307 (2010).
37. Silvis-Cividjian, N., Hagen, C., Leunissen, L. & Kruit, P. The role of secondary electrons in electron-beam-induced-deposition spatial resolution. Microelectron. Eng. 61, 693 (2002).
38. Winston, D. et al. Scanning-helium-ion-beam lithography with hydrogen silsesquioxane resist. J. Vac. Sci. Technol. B 27, 2702 (2009).

Acknowledgements

We would like to thank Mr. M. Komatsu of the National Institute for Materials Science (NIMS) for fabricating the in situ heating sample holder. We appreciate Ms. S. Nagano for the supports of HIM characterization. This work is in part supported by MEXT Program for Development of Environmental Technology using Nanotechnology.

Author Contributions

H.G. and D.F. planned the experiments. K.M. prepared the FNW sample. H.G. and C.W. performed the experiments, interpreted the results and wrote the paper. H.W. and H.M. performed the Raman characterization. All authors reviewed the manuscript.

Additional Information

Competing financial interests: The authors declare no competing financial interests.

How to cite this article: Guo, H. et al. Thermal decomposition of fullerene nanowhiskers protected by amorphous carbon mask. Sci. Rep. 6, 38760; doi: 10.1038/srep38760 (2016).

Publisher’s note: Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

This work is licensed under a Creative Commons Attribution 4.0 International License. The images or other third party material in this article are included in the article’s Creative Commons license, unless indicated otherwise in the credit line; if the material is not included under the Creative Commons license, users will need to obtain permission from the license holder to reproduce the material. To view a copy of this license, visit http://creativecommons.org/licenses/by/4.0/

© The Author(s) 2016