Shah, Ali; Stenberg, Petri; Karvonen, Lasse; Ali, Rizwan; Honkanen, Seppo; Lipsanen, Harri; Peyghambarian, N.; Kuittinen, Markku; Svirko, Yuri; Kaplas, Tommi

**Pyrolytic carbon coated black silicon**

*Published in:*
Scientific Reports

*DOI:*
10.1038/srep25922

*Published: 13/05/2016*

*Document Version*
Publisher's PDF, also known as Version of record

*Published under the following license:*
CC BY

*Please cite the original version:*
Shah, A., Stenberg, P., Karvonen, L., Ali, R., Honkanen, S., Lipsanen, H., Peyghambarian, N., Kuittinen, M., Svirko, Y., & Kaplas, T. (2016). Pyrolytic carbon coated black silicon. *Scientific Reports, 6*, [25922]. https://doi.org/10.1038/srep25922

This material is protected by copyright and other intellectual property rights, and duplication or sale of all or part of any of the repository collections is not permitted, except that material may be duplicated by you for your research use or educational purposes in electronic or print form. You must obtain permission for any other use. Electronic or print copies may not be offered, whether for sale or otherwise to anyone who is not an authorised user.
Pyrolytic carbon coated black silicon

Ali Shah1, Petri Stenberg2, Lasse Karvonen1, Rizwan Ali1, Seppo Honkanen2, Harri Lipsanen1, N. Peyghambarian1,2,3, Markku Kuittinen2, Yuri Svirko2 & Tommi Kaplas2

Carbon is the most well-known black material in the history of man. Throughout the centuries, carbon has been used as a black material for paintings, camouflage, and optics. Although, the techniques to make other black surfaces have evolved and become more sophisticated with time, carbon still remains one of the best black materials. Another well-known black surface is black silicon, reflecting less than 0.5% of incident light in visible spectral range but becomes a highly reflecting surface in wavelengths above 1000 nm. On the other hand, carbon absorbs at those and longer wavelengths. Thus, it is possible to combine black silicon with carbon to create an artificial material with very low reflectivity over a wide spectral range. Here we report our results on coating conformally black silicon substrate with amorphous pyrolytic carbon. We present a superior black surface with reflectance of light less than 0.5% in the spectral range of 350 nm to 2000 nm.

Creating a superior black surface has long been a lasting challenge for many decades1,2. Most promising results to-date in the visible (VIS) and ultra-violet (UV) spectral ranges have been achieved with black silicon (bSi), a silicon surface decorated with randomly distributed vertical pyramidal needles of micrometer size height and a very large aspect ratio1,2. Photons incident on the bSi surface experience multiple reflections on pyramids’ surfaces and photons with energies higher than the silicon bandgap are rapidly absorbed by silicon. Such a surface structure provides superior suppression of the reflectivity by three orders of magnitude in the VIS, making bSi a versatile platform for highly effective solar cells3,4 or THz emission sources5. However, the anti-reflectance performance of bSi reduces rapidly when the incident photon energy becomes lower than the silicon bandgap, i.e. the multiple reflections cannot provide the required energy loss5. In order to extend the absorbing range towards mid- and far-IR range the pyramidal needles that compose the bSi surface should be able to strongly absorb such photons while having the same aspect ratio and spatial distribution. In order to achieve that goal, a modification of bSi and deposition of broadband absorbing materials on bSi have been proposed6,7. Very recently it was observed that highly-doped bSi coated with a subwavelength thick oxide layer using atomic layer deposition technique is also capable to suppress reflectivity over a wide spectral range8,9. One may expect that an ultra-thin, strongly absorbing carbon layer which is conformally deposited on bSi can also be used to enhance its antireflection properties.

Among carbon based materials, sp2 hydrogenated carbon is well known for its absorption in a wide spectral range from UV to far IR due to the π-band electronic absorption. Quite recently, carbon nanotubes (CNTs) and graphene have demonstrated great capability to efficiently absorb electromagnetic radiation and have been suggested as perfect black materials10,11. However, CNTs and graphene pose serious challenges in fabrication since both of these materials typically require a catalyst substrate for the growth. Also, the deposition of these materials on nano- and microstructured surfaces like bSi is difficult task that makes their use in practical devices questionable.

Alternative route to improve bSi performance is to employ pyrolytic carbon (PyC), an amorphous carbon allotrope with dominating sp2 bonding. PyC - similar to carbon nanotubes (CNTs) and graphene - displays strong absorption of the EM radiation in a wide spectral range spanning from UV to far-IR10,12. It is conventionally produced by chemical vapor deposition (CVD) from gaseous hydrocarbon precursors at about 1000 °C14,15. The deposited PyC film consists of small, lamellar graphitic ribbons with size of several nanometers and relatively small amount of amorphous carbon16,17. Due to the dominating sp2 hydrogenation and π-electrons, it is expected to have strong optical absorption over a wide spectral band18. This property makes PyC films attractive for designing new types of black surfaces that can complement materials such as black silicon (bSi).

1Department of Micro and Nanosciences, Aalto University, Espoo, P.O. Box 13500, FI-00076, Finland. 2Institute of Photonics, University of Eastern Finland, Joensuu, P.O. Box 111, FI-80101, Finland. 3College of Optical Sciences, University of Arizona, 1630 E. University Blvd, Tucson, AZ 85721, USA. Correspondence and requests for materials should be addressed to T.K. (email: tommi.kaplas@uef.fi)
It has been recently demonstrated that CVD with light hydrocarbons (e.g. acetylene and methane) can be employed not only for fabrication of carbon fiber-reinforced carbon but also for deposition of the nanometer thin PyC films on dielectric substrates. This experimental finding has opened a wide range of new applications of PyC including electrochemical measurements, conductive vias through dielectric, carbon-silicon Schottky-barrier diodes, and shielding of the electromagnetic radiation in a wide spectral range.

Despite the fact that amorphous carbon inclusions make electrical conductivity of PyC films somewhat lower than that of graphene, the transmittance of a PyC film resembles that of graphene. In particular, the transmittance of PyC is near constant in the near-IR spectral range and a wide absorption peak is located at 270 nm, at the same wavelength as for graphene. Moreover, catalyst free synthesis of PyC permits coating of metal and dielectric substrates of any shape.

In this paper, we demonstrate a novel superior black surface for the spectral range of UV to mid-IR that employs PyC-enhanced bSi structure. By developing an inexpensive and scalable technique we achieved conformally coated bSi surface with the PyC film thickness of 25 nm and demonstrated that such coating enormously suppresses reflectance in the IR. The obtained results allow us also to revisit the long-standing discussion on the mechanism of the CVD of PyC from light hydrocarbons. In particular, the demonstrated conformal growth on micro- and nanostructured substrates indicates that the surface activated growth process is the dominant mechanism of the PyC film synthesis.

For the demonstration of ultra-thin PyC formation on Si, we deposit 25 nm thick layer of PyC on a bare Si substrate. Figure 1a,b show the SEM images of the PyC layer grown on a bare Si substrate. The film grows uniformly all around the substrate covering the sample surface with a thin layer of PyC. Although, the PyC layer is only 25 nm thick, it has crucial impact on the reflectance of the Si. Specifically, one can observe from Fig. 1c that the deposition of ultra-thin PyC layer makes the reflectance nearly wavelength independent in the range from 400 nm to 2000 nm. In UV, however, the reflectance decreases rapidly due to the strong absorption in the PyC film. This absorption resembles the M-saddle point absorption of graphene and has its maximum around the same wavelength as for graphene. Moreover, catalyst free synthesis of PyC permits coating of metal and dielectric substrates of any shape.

In the characterization of PyC with Raman measurements, the disorder induced D peak and graphitic G mode are conventionally used to characterize the crystallinity of the fabricated film. The measured Raman spectrum shown in Fig. 1d (and Fig. S1, Supplementary information) has a weak D' peak at around 1500–1550 cm⁻¹, which is a signature of the amorphous carbon.

Moreover, in contrast to Raman spectrum of highly crystalline graphite and graphene, the 2D' peak in PyC film, shown in Fig. 1d is greatly widened and significantly suppressed. Analysis of the Raman spectra
(Supplementary information) allows us to conclude that in the deposited PyC film, the average in-plane lattice parameter is around 2.2 nm. The exceptional conformality of the PyC coating facilitates deposition on nanostructured surfaces with sharp and randomly distributed surface features. In Fig 2a,c, one can observe a bSi without and with PyC, respectively. Similar to Si gratings (see Supplementary information), the PyC film covers the pyramidal silicon features with exceptional conformality offering negligible film surface roughness. In Fig. 2c one may observe that the PyC coating does not produce visible changes in bSi structure. This obtained high quality conformal PyC coating indicates that the surface of the PyC film remains active throughout the whole CVD process. Specifically, the PyC film keeps growing uniformly despite a few tens of nanometers gap between the nanostructures, see Fig. 2d (and Fig. S2 in the Supplementary information). This is a clear signature of dominating role of the reactive surface in the PyC deposition that leads to the highly uniform and conformal coating of the patterned substrate.

The reflectance of bSi after PyC coating, shown in Fig. 3a, changes drastically in the IR ($\lambda > 1000$ nm). Specifically, while bare bSi reflects more than 80% of the incident light from 1200 nm to 2000 nm (Fig. 3a), the reflection of bSi coated with PyC film is as low as 0.5% (Fig. 3c). With the wavelength higher than 2000 nm, the reflectance of thin PyC film coated bSi increases as the bSi microstructures are not deep enough for the corresponding wavelengths (see Supplementary information). It is worth noting that, although bare Si has lower reflectance compared to PyC coated Si in the VIS spectral range (Fig. 1), the anti-reflecting properties of PyC coated bSi in VIS-IR spectral range is superior to that of bare bSi (Fig. 3b).

The strong IR absorption of PyC coated bSi is very similar to the results obtained by using strongly doped silicon, but the absorption mechanism is very different. The strong doping increases the number of free electrons in silicon and thus increases light absorption inside the silicon substrate. Since the bSi surface is acting as an anti-reflecting surface, the light is trapped and absorbed by the strongly doped silicon substrate. At light doping, the silicon itself does not absorb as seen from the reflectance spectrum in Fig. 3a. Instead, the absorbing carbon layer, which is only 25 nm thick, interacts with the incident light and absorbs it at the surface of the bSi substrate. Thus, the performance of the PyC-coated structure is actually not dependent on the absorption of the substrate.

Since the IR absorption in lightly doped silicon is relatively weak, the strong reflection at near IR originates from the backside reflection of the bare and black silicon substrate. In order to demonstrate this, we measured reflectance of black and bare Si with a spectron disk as a backside reflector. This configuration highlights the influence of the 25 nm thick PyC film on the reflectivity. Specifically, we showed that both bare and black Si reflect about 80% and 50% of the incident IR radiation with and without backside reflector, respectively. However, the reflectance of the PyC coated samples measured with or without the spectron disk was the same, which is an evidence of very strong IR radiation absorption in the ultra-thin PyC film. Although, in our experiment we did not measure the dependence of the reflectivity on the angle of incidence, we believe that the PyC coating should suppress reflection also at oblique incidence. This is because 25 nm thick PyC layer is thin enough not to affect to a few micron deep features on the bSi surface.

Since the in-plane size of intertwined graphene nanoribbons in PyC is about a few nanometers, strong electron scattering makes the electrical conductivity of PyC lower than that of graphene and also suppresses electronic and nonlinear-optical effects. Thus one may expect that absorbance of graphene coated bSi will outperform
PyC coated bSi. However, coating of the pyramidal bSi surface conformally by graphene is hardly achievable. Furthermore, monolayer thick graphene is not enough for sufficient absorbance. Meanwhile, PyC offers a reliable conformal coating method as 25 nm thick PyC layer doesn’t make changes to the structural appearance of bSi.

The conformality of PyC is attributed to the nature of the CVD process, where the active sites are randomly distributed over the surface. Hydrocarbon molecules on these active sites are absorbed followed by hydrogen desorption. Thus, one may expect that this mechanism can result in synthesis of the solid PyC layer that will precisely follow the shape of a substrate. In the initial stage of the CVD process, the methane decomposes forming carbon dimers (C2) in gaseous atmosphere. Interaction of dimers with other gaseous species in the CVD chamber leads to the formation of aromatic C6-based compounds by C2→C6 route (other routes are also possible) that attach on the substrate forming new active sites for substrate-gas interaction. Specifically, when a massive aromatic species land on the substrate, they may continue evolving thus forming active surface sites that initiate growth of the carbon rings both at the edges and on the top of basal planes. Such surface activated synthesis explains the conformal growth of PyC over the arbitrary shaped substrate surface.

In summary, we have demonstrated a wide band anti-reflecting surface by combining bSi with PyC thin film. The structural integrity of the substrate was not compromised and enhanced light absorption was demonstrated over a wide band. Absorption and Raman spectroscopic measurements confirm that the PyC film is uniform over the substrate and the structure is amorphous. One may expect a lower DC conductivity and optical nonlinearity compared with highly oriented graphite films and graphene. However, the excellent uniformity and extreme conformality of PyC films over large areas indicates dominant heterogeneous growth mechanism. The advantages of PyC coating make it preferable and practical for many potential applications. In particular, biocompatibility and chemical inertness of PyC coated bSi suggest use in biomedical sensors and increase the sensitivity in e.g. electrochemical experiments, where the molecule adhesion plays a crucial role. Moreover, since an ultra-thin PyC film is robust and easily holds its weight, it could be used as a membrane material for a free-standing gratings in e.g. micro- and nanoelectromechanical systems.

Methods

Sample preparation. In order to demonstrate the conformal PyC deposition, we coated a bare silicon substrate and bSi with PyC films. The bSi was formed in a cryogenic, inductively coupled plasma reactive ion etcher (ICP-RIE; Plasmalab System 100, Oxford Instruments, UK) using the following optimized parameters: 40 sccm SF6, 18 sccm O2, 6 W forward power, 1000 W ICP power, −110 °C, 10 mTorr pressure and helium backside.
cooling. Details of bSi fabrication are reported in previous studies\(^{29,30}\). The silicon used in the experiment was lightly p-doped, 0.5 mm thick silicon wafers.

The PyC coatings of the bare Si and bSi were performed by using a hot wall CVD system described in ref. 10. The system employs methane/hydrogen mixture as a carbon source and is equipped with a cylindrical quartz chamber, a tubular furnace (Carbolite CTF 12/75/700) and a computer controlled vacuum pump and gas flow controllers. The thickness of the fabricated PyC films is determined by the methane concentration in the hydrogen-methane gas mixture and pressure in the CVD chamber.

Before heating the oven, the CVD chamber was purged twice with nitrogen and once with hydrogen. In the first stage of the process, the chamber was heated up to 700 °C during one hour in hydrogen atmosphere (7 mbar). At 700 °C, it was pumped down to vacuum and the hydrogen was replaced with hydrogen-methane gas mixture. After the gas exchange the chamber was heated from 700 °C to 1100 °C in 40 min. At 1100 °C, the catalyst-free spontaneous methane decomposition starts. This process is initiated by methane radicals that form new hydrocarbon molecules, which are combined into massive polycyclic aromatic structures and deposited onto substrate\(^{27,28}\). After the process both sides of the substrate were coated with the PyC film. In our experiments we did not observe any dependence on the sample position inside the chamber, i.e. PyC films of the same thickness were deposited on the both sides of the horizontally and vertically oriented substrates. The gravity is not involved in the deposition process, i.e. it is unlikely dominated by landing of massive hydrocarbon particles on the substrates. The dependence of the deposited PyC film thickness on methane concentration can be found in ref. 17.

In order to demonstrate that developed technique can be employed for PyC deposition on arbitrary substrates, we also coated one-dimensional gratings fabricated from silica, silicon and TiO\(_2\) substrates (see Supplementary information).

**Characterization.** The optical reflectance characterization for PyC was done by using spectrophotometer (Perkin Elmer lambda 1050 with 8/d geometry using spectral range from 350 nm to 2000 nm. Measurement was done using a spectralon disk as a backside mirror and about 1 mm \(\times\) 10 mm beam spot size. Raman spectroscopy was done by Renishaw inVia Raman microscope with excitation wavelength of 514 nm. The excitation beam intensity was kept low in order to avoid heat induced phenomena in the carbon film. Scanning electron microscopy was done by SEM Zeiss Supra 40 and Leo 1550 Gemini.

**References**

1. Gittleman, J. I., Sichel, E. K., Lehmann, H. W. & Widmer, R. Textured silicon: A selective absorber for solar thermal conversion. *Applied Physics Letters* 35, 742–744 (1979).
2. Chattopadhyay, S. et al. Anti-reflecting and photonic nanostructures. *Materials Science and Engineering B* 89, 1–35 (2010).
3. Oh, J., Yuan, H.-C. & Branz, H. M. An 18.2%-efficient black-silicon solar cell achieved through control of carrier recombination in nanostructures. *Nature Nanotechnology* 7, 743–748 (2012).
4. Savin, H. et al. Black silicon solar cells with interdigitated back-contacts achieve 22.1% efficiency. *Nature Nanotechnology* 10, 624–628 (2015).
5. Hoyer, P., Theuer, M., Beigang, R. & Kley, E. B. Terahertz emission from black silicon. *Appl. Phys. Lett.* 93, 091106 (2008).
6. Wang, Y., Liu, Y. P., Liang, H. L., Mei, Z. X. & Duab, X. L. Broadband antireflection on the silicon surface realized by Ag nanoparticle-patterned black silicon. *Phys. Chem. Chem. Phys.* 15, 2345–2350 (2013).
7. Yang, J. et al. Design and fabrication of broadband ultralow reflectivity black Si surfaces by laser micro/nanoprocessing. *Light: Science & Applications* 3, 1–8 (2014).
8. Otto, M. et al. Conformal Transparent Conducting Oxides on Black Silicon. *Adv. Mater.* 22, 5035–5038 (2010).
9. Steglich, M. et al. An ultra-black silicon absorber. *Laser Photonics Rev.* 8(2), L13–L17 (2014).
10. Mizio, K. et al. A black body absorber from vertically aligned single-walled carbon nanotubes. *PNAS* 106(15), 6044–6047 (2009).
11. Alaea, K., Farhat, M., Rockstuhl, C. & Lederer, F. A perfect absorber made of a graphene micro-ribbon metamaterial. *Opt. Express* 20, 28017–28024 (2012).
12. Charlebois, F. et al. Y. Nonlinear refraction in semitransparent pyrolytic carbon films. *Opt. Mater. Express 2, 1822–1827 (2012).
13. Batrakov, K. et al. Enhanced microwave shielding effectiveness of ultrathin pyrolytic carbon films. *Appl. Phys. Lett.* 103, 073117 (2013).
14. Benzeninger, W., Becker, A. & Hüttner, K. J. Chemistry and kinetics of chemical vapour deposition of pyrocarbon: I. Fundamentals of kinetics and chemical reaction engineering. *Carbon* 34, 957–966 (1996).
15. Pierson, H. O. *Handbook of carbon, graphite, diamond and fullerences* Vol 1 (ed. Pierson, H. O.) Ch. 7, 141–164 (Noyes Publications, 1993).
16. McEvoy, N. et al. Synthesis and analysis of thin conducting pyrolytic carbon films. *Carbon* 50, 1216–12126 (2012).
17. Kaplas, T. & Svirko, Y. Direct deposition of semitransparent conducting pyrolytic carbon films. *J. of Nanophot.* 61(1), 061703–6 (2012).
18. Taft, E. A. & Philipp, H. R. Optical Properties of Graphite. *Phys. Rev.* 138, A197–A220 (1965).
19. Graham, A. P., Schindler, G., Duesberg, G. S., Lutz, T. & Weber, W. An investigation of the electrical properties of pyrolytic carbon in reduced dimensions: Vias and wires. *J. of Appl. Phys.* 107, 114316 (2010).
20. Yin, C., McEvoy, N., Rezvani, E., Kumar, S. & Duesberg & G. S. Carbon–Silicon Schottky Barrier Diodes. *Small* 8(9), 1360–1364 (2012).
21. Bae, S. et al. Roll-to-roll production of 30-inch graphene films for transparent electrodes. *Adv. Mater.* 22, 742–744 (2010).
22. Mak, K. F., Ju, L., Wang, F. & Heinz, T. F. Optical spectroscopy of graphene: From the far infrared to the ultraviolet. *Solid State Communications* 152, 1341–1349 (2012).
23. Reich, S. & Thomsen, C. Raman spectroscopy of graphite. *Phys. Trans. R. Soc. Lond. A* 362, 2271–2288 (2004).
24. Ferrari, A. C. & Robertson, J. Interpretation of Raman spectra of disordered and amorphous carbon *Phys. Rev. B* 61(20), 14095–13 (2000).
25. Jawhari, T., Roid, A. & Casado, J. Raman spectroscopic characterization of some commercially available carbon black materials. *Carbon* 33(11), 1613–22 (1995).
26. Oberlin, A. Pyrocarbons. *Carbon* 40(1), 7–24 (2002).
27. Becker, A. & Hüttner, K. J. Chemistry and kinetics of chemical vapor deposition of pyrocarbon - IV pyrocarbon deposition from methane in the low temperature regime. *Carbon* 36, 213–224 (1998).
28. Hubner, S., Miyakawa, N., Pahlke, A. & Kreupl, F. Performance Improvement of Graphenic. *Carbon X-ray Transmission Windows*, *MRS Advances* 14th March 1–6 (2016).
29. Sainiemi, L. et al. Non-reflecting silicon and polymer surfaces by plasma etching and replication. *Adv. Mater* 23(11), 122–6 (2011).
30. Shah, A., Chang, B., Suhkonen, S., Zhou, Q. & Lipsanen, H. Surface-Tension-Driven Self-Alignment of Microchips on Black-Silicon-Based Hybrid Template in Ambient Air. *J. Microelectromechanical Syst.* 22(3), 739–746 (2013).
Acknowledgements
The research leading to these results has received funding from the European Union Seventh Framework Programme under grant agreement no. 604391 Graphene Flagship, NP-Nano FiDiPro-project by the Finnish Funding agency for Technology and Innovation (TEKES), Academy of Finland project no. 287886. AS and HL acknowledge the support from Aalto Energy Efficiency project Moppi. Support from the US National Science Foundation is also acknowledged through the CIAN ERC.

Author Contributions
A.S. fabricated bSi samples; T.K. made the PyC coatings; A.S. and T.K. performed SEM imaging and reflectance measurements of bSi samples; T.K. performed Raman measurements and data analyzing; P.S. fabricated grating structures and SEM imaging of PyC coated samples (Supp. info); all co-authors participated in development and discussions; T.K., A.S. and Y.S. co-wrote the paper.

Additional Information
Supplementary information accompanies this paper at http://www.nature.com/srep

Competing financial interests: The authors declare no competing financial interests.

How to cite this article: Shah, A. et al. Pyrolytic carbon coated black silicon. Sci. Rep. 6, 25922; doi: 10.1038/srep25922 (2016).

This work is licensed under a Creative Commons Attribution 4.0 International License. The images or other third party material in this article are included in the article’s Creative Commons license, unless indicated otherwise in the credit line; if the material is not included under the Creative Commons license, users will need to obtain permission from the license holder to reproduce the material. To view a copy of this license, visit http://creativecommons.org/licenses/by/4.0/