Effects of NP-SR Fertilizer Composition and Water Logging on Soil Chemical Properties and N Fertilizer Efficiency in Paddy Field

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ABSTRACT
Rice is the primary food commodity in Indonesia. To increase the rice production, urea fertilizer has been excessively used, specifically on marginal land. However, it has no desired effect on the plant’s Nitrogen uptake due to volatilization. Previous studies suggest the use of zeolite to be mixed with urea to reduce the volatilization rate. This study aimed to determine the effect of six NP-SR (Nitrogen Phosphorus Slow Release) fertilizer compositions (without NP-SR, 20.69-0; 19.7-5; 17.27-10; 15.9-15; and 18.94-20) on soil chemical properties, nitrogen efficiency, and paddy rice yield on Ultisols at three waterlogging levels (0.5 cm, 3 cm, and 5 cm). The study was an experiment arranged in a Randomized Complete Block Design (RCBD) consisting of two factors, which were waterlogging levels and NP-SR fertilizer compositions. The results showed that waterlogging treatments significantly affected soil chemical properties in the initial vegetative and harvest phases. It can be concluded that the treatment of flooding (up to 3 cm) and fertilizer (NP-SR of 15.90-15) application could significantly reduce the volatilization rate, increase the efficiency of N fertilizer, and increase rice yield.

Keywords: N efficiency, Paddy, Ultisol, Zeolite

INTRODUCTION
Rice is the main food commodity in supporting the needs of the Indonesian people (Amrullah, 2018). Rice plants (Oryza sativa L.) have become essential food crops, which are the staple food of more than half of the world’s population. However, there have been some problems in efforts to increase rice crop production, including the increasing environmental damage and global climate change, limited infrastructure availability, and increasing conversion from agricultural land to non-agricultural land (Wahyuni et al., 2015).

The increase in national rice production could be achieved through the extensification of agricultural land outside Java. Nevertheless, islands over Indonesia apart from Java are dominated with acidic mineral soil, such as ultisol. Without proper management, this soil inhibits plant or food crops growth due to low pH, low organic matter, low organism and biodiversity, and low productivity. Also, nutrients in ultisol are generally low due to intensive washing, while the low organic matter is caused by rapid decomposition and erosion (Syahputra et al., 2015). Thus, this type of soil should be treated with a large amount of fertilizer and...
ameliorant (Aainaa et al., 2014).

Syahputra et al. (2015) reported that the low total nitrogen (N) content in the six of the Ultisol sub-groups (Typic Hapludults, Typic Paleudults, Psammentic Paleudults, Typic Plinthudults, Typic Ochraquults, and Typic Paleaquults) was due to the low content of soil organic carbon, loss due to washing, and evaporation. Plants absorb N in nitrate ions because there has been a change in the form of $\text{NH}_4^+$ to $\text{NO}_3^-$. In flooded condition, plants absorb N in the form of $\text{NH}_4^+$ (Han et al., 2016). Nitrogen is a very mobile nutrient, rapidly experiencing volatilization, which releases $\text{NH}_3$ into the air and loses N through leaching (Cai et al., 2002). Most N losses occur because of the volatilization of $\text{NH}_3$ (Li et al., 2017). N losses range from 10 to 50% of N fertilizer applied in rice cultivation (Coskun et al., 2017). N loss through the volatilization of $\text{NH}_3$ is very significant, mainly if N fertilizer is applied through dispersion, which can reach 50% (Sommer et al., 2004).

Nitrogen losses are also influenced by the rice fields’ water condition, which will affect the oxidation and reduction processes in the paddy field, affecting the amount of $\text{NH}_3$ volatile gas (Watanabe et al., 2009). In flooded condition, soil pH affects the volatilization process rate (Hadjowigeno & Rayes, 2005) so that the management of flooding is expected to reduce N loss. The efficient use of fertilizers is an essential part of intensive rice farming systems. Chemical fertilizer use efficiency by farmers is currently less than 60% (Sudarman 1990). Inefficient use of N fertilizers is caused by N loss from the soil system through evaporation in the form of ammonia, experiencing denitrification, erosion, and protection (Timilsena et al., 2015).

Increasing fertilizer efficiency can be done by improving cultivation and irrigation techniques, fertilization techniques, and fertilizer properties. Through these efforts, it is expected that the release of nutrients can be more regulated according to plants' needs so that nutrient losses outside the soil system can be reduced, and pollution to the environment can be decreased (Sastiono, 2004).

The use of slow-release and controlled urea fertilizers has been shown to reduce N loss in rice cultivation and improve rice yields (Ye et al., 2013; Li et al., 2018). One way to improve fertilizer efficiency is to mix fertilizer with zeolite (Suwardi, 2009). The use of natural zeolite can reduce the volatilization rate of $\text{NH}_3$ from N fertilizer because this mineral has ample pore space to absorb and exchange cations (Van Straaten, 2002; Noori et al., 2006). This mechanism is the basis of this research in the use of zeolite as a mixture in urea.

Previous research showed that mixing urea with zeolite could reduce volatilization up to 6-44% compared to urea without zeolite (Ahmed et al., 2010; Palanivell et al., 2015). The use of natural nutrient-enriched zeolite N can continuously improve chemical properties, thereby improving soil fertility. Efforts to improve soil fertility can be achieved by providing NPSR fertilizer so that it is expected to have an optimal influence on agricultural cultivation in acidic mineral soils. This study aimed to determine the effect of various NPSR fertilizer compositions on soil chemical properties, nitrogen efficiency, and paddy rice yield on Ultisols at various waterlogging.

**MATERIALS AND METHODS**

**Experimental design**

The study consisted of two stages. The first stage was the acidulation of Natural Rocks Phosphate (NRP), NPSR fertilizer production and greenhouses experiments. The second stage was testing NPSR fertilizers on various compositions and various waterlogging levels using paddy plants (Inpari 32) on Ultisol’s soil order.

The NPSR fertilizer was made from urea, natural zeolite, natural phosphate rock, humid acid, and vertisol adhesive. The zeolit used has a pH
H₂O of 7.35 and a pH KCl of 6.20 with a DHL value of 151 µS per cm. The SiO₂, Al₂O₃, and CaO content of the zeolite were 75.8%, 15.4%, and 2.45%, respectively. Humic and sulfat acids are obtained from compost extracted with NaOH, and then used for hydrothermal acidulation of natural phosphate rock.

The experiment of this study was arranged in a Randomized Complete Block Design (RCBD), consisting of two factors. The first factor was waterlogging level, consisting of W0 (0.5 cm depth), W1 (3 cm), and W2 (5 cm). Meanwhile, the second factor was NP-SR fertilizer composition level, consisting of F0 (without NP-SR fertilizer application), F1 (provision of NP-SR fertilizer with a composition of 20.69% - 0%), F2 (19.70% - 5%), F3 (17.27% - 10%), F4 (15.90% - 15%), and F5 (18.94% - 20%). Therefore, there were 18 treatment combinations with three replications in each treatment, resulting in 54 experimental units. The dose of fertilizer used was equivalent to 400 kg N/ha, which examined from April 2018 to February 2019 at the Screen House of the Faculty of Agriculture, Jenderal Soedirman University.

Soil Analysis

Soil sampling for analysis of soil chemical properties was carried out three times, 0 days after transplanting (DAT), the final vegetative phase of rice (75 DAT), and harvest (125 DAT). Soil samples with the Ultisol order were taken from Tanggeran, Banyumas, Central Java. Samples were taken at a depth of 0-30 cm, dried, crushed, and filtered with a 2 mm sieve. Before treatment, soil samples were incubated for one week. Observation of NH₃ gas was carried out every week during the experiment until harvest.

The measurement of NH₃ gas evaporation (ammonia) was carried out by titrating the gas contained in 0.1 N H₂SO₄ solution once a week starting from one week after transplanting until harvest. N was calculated using the following equations:

1. N absorbed (mg N/chamber) = (volume of H₂SO₄ 0.1 N container - volume of NaOH 0.1 N used for titration) x 0.1 x 14
2. N volatilization (mg N/chamber) = N volatilization in fertilizer - N volatilization in control
3. N volatilization (%) = (N volatilization (mg N/chamber))/(N in fertilizer (mg N/chamber)) x 100 (Wang et al., 2004)
4. Fertilizer efficiency (%) = 100 - %N volatilizations

Statistical Analysis

The data were analyzed using ANOVA (Analysis of Variance), followed with Duncan’s Multiple Range Test at 5% to determine the differences between treatments using Excel spreadsheet and DSSTAT software. The treatment that showed the significant highest value based on DMRT was considered the best treatment.

Table 1. Chemical properties of Zeolites

| Parameter    | Average |
|--------------|---------|
| pH H₂O (1:5) | 7.35    |
| pH KCl (1:5) | 6.20    |
| EC (1:5) µS.cm⁻¹ | 151    |
| SiO₂ (%)    | 75.8    |
| Al₂O₃ (%)   | 15.4    |
| CaO (%)     | 2.45    |

Remarks: Analysis conducted at the Laboratory of Geological Engineering, Jenderal Soedirman University.

RESULTS AND DISCUSSION

The primary component in NP-SR fertilizer is zeolite. Table 1 presents the results of the laboratory analysis of the chemical properties of zeolites. Zeolites used in this study had pH H₂O and pH KCl of 7.35 and 6.20, respectively, with electrical conductivity (EC) value of 151 µS cm⁻¹, and the content of SiO₂, Al₂O₃, and CaO of 75.8, 15.4, and 2.45%, consecutively. The zeolite pH value in this study is high, which is similar to the pH value
of the zeolite used by Bundan et al. (2011) with pH H₂O of 7.90. The silica content of zeolite used in this study (73.2%) is almost the same as the zeolite from Gunung Kidul (Safrihatini, 2017). Various metal elements found in zeolite cause zeolite to be applied as a cation exchanger because zeolite contains various metal elements that could be exchanged with other desired metals.

The analysis of variance on the growth variables showed that the application of NPSR fertilizer composition had a significant effect on the plant height and leaf area, while water logging level had a significant effect on the leaf area only (Table 2). A similar thing was reported by Juniadi et al. (2017) that rice planted in stagnant conditions with a water level of 3-10 cm did not affect the plant height. There was no significant interaction effect of NPSR fertilizer and watering on the growth component.

The treatment of NPSR fertilizer composition can significantly increase plant height, and this indicates that plant growth is strongly influenced by the presence of nutrients (Kavoosi, 2007). The difference in the plant height in each treatment is influenced by the nutrient content contained in the soil. N nutrient, in this case, is closely related to photosynthesis. Nitrogen is useful for accelerating growth of vegetative parts of plants. N nutrient deficiency causes dwarf plants (small), small tillers, and narrow and pale-yellow leaves (Taiz & Zeiger, 1991).

Table 2. The results of the analysis of variance on plant growth, soil chemical properties, plant N uptake, yields of rice, and NH₃ evaporation as affected by the application of NPSR fertilizer composition and waterlogging level

| Variable                              | Treatment | W | F   | W x F |
|---------------------------------------|-----------|---|-----|-------|
| **Growth variables**                  |           |   |     |       |
| Plant height                          |           | ns| **  | ns    |
| Leaf area                             |           | **| **  | ns    |
| **Soil chemical properties**          |           |   |     |       |
| N-total 0 days after transplanting (DAT) |           | ns| *   | ns    |
| N-total 75 DAT                        |           | ns| ns  | *     |
| N-total 125 DAT                       |           | ns| ns  | **    |
| pH H₂O 0 DAT                          |           | * | ns  | ns    |
| pH H₂O 75 DAT                         |           | ns| ns  | ns    |
| pH H₂O 125 DAT                        |           | * | ns  | ns    |
| pH KCl 0 DAT                          |           | * | ns  | ns    |
| pH KCl 75 DAT                         |           | ns| ns  | ns    |
| pH KCl 125 DAT                        |           | ns| ns  | ns    |
| EC 0 DAT                              |           | ns| ns  | ns    |
| EC 75 DAT                             |           | ns| ns  | ns    |
| EC 125 DAT                            |           | ns| ns  | ns    |
| **Yields of rice**                    |           |   |     |       |
| Percentage of filled grains           |           | ns| **  | **    |
| Percentage of empty grains            |           | ns| **  | **    |
| NH₃ evaporation                       |           | **| **  | ns    |
| **N fertilizer efficiency**           |           |   |     |       |

Remarks: ns = not significantly different, * = significant, ** = very significant, F = NPSR fertilizer composition, W = water logging level, W x F = interaction between water logging level and NPSR fertilizer composition.
The high level of waterlogging showed a very significant effect on the leaf area. The amount of nutrients absorbed by plant roots depends on the amount of moisture in the soil, determined by soil particles’ ability to hold water and the roots’ ability to absorb it. According to Gardner et al. (1991), lack of water in the vegetative phase causes leaves shrinkage, stem growth depression, and an increase in abscisic acid (ABA) stimulating stomatal closure, which results in reduced assimilation of CO$_2$ to the leaves, thereby inhibiting the plant growth.

There was interaction effect of the combination of NP-SR fertilizer composition and the of water level on the total N content of the soil at 75 DAT and 125 DAT (Table 3). From the comparison of total soil N values before applying zeolite, which is 0.08% to 0.21%, it can be concluded that the administration of NP-SR fertilizer composition can increase the total N value of the soil. This condition occurred because zeolite is one of the ingredients that can temporarily bind Nitrogen. Zeolites have a high cation exchange capacity (CEC) (between 120-180 cmol (+) kg$^{-1}$), which is useful for adsorption, binding, and cation exchange (He et al., 2002).

Table 3 shows that at 75 DAT, the application of NPSR fertilizer with a composition of 19.70-5 (F2) at the waterlogging of 0.5 cm increased the soil’s total N content by 21.43% compared to without fertilization. The application of NPSR fertilizer with a composition of 15.90-15 (F4) and 18.94-20 (F5) at the waterlogging of 0.5 cm gave the same results for the soil’s total N content, which was 0.15%, considered low. Meanwhile, the application of NPSR fertilizer with a composition of 15.90-15 (F4) at the waterlogging level of 3 cm at 75 DAT increased the total N content of the soil by 50% compared to without fertilization. The soil’s total N content is affected by waterlogging because water determines nutrient translocation related to oxidation and reduction reactions in solution.

The actual pH is generally higher than the potential pH. Several processes in the soil are affected by soil reactions. The analysis of pH H$_2$O, pH KCl, and soil EC at 0 days after transplanting

| Observation            | Composition of NP-SR fertilizer (K) | Waterlogging (W) | Mean of N-total (%) |
|------------------------|-------------------------------------|------------------|---------------------|
|                        |                                     | W0 (0.5 cm)      | W1 (3 cm)           | W2 (5 cm)           |
| 75 days after transplanting | F0 (without fertilizer)         | 0.13 d B         | 0.14 f A            | 0.14 d A            | 0.14                |
|                        | F1 (20.69-0)                      | 0.13 d C         | 0.15 e B            | 0.17 b A            | 0.15                |
|                        | F2 (19.70-5)                      | 0.17 a A         | 0.16 d B            | 0.17 b A            | 0.16                |
|                        | F3 (17.27-10)                     | 0.14 c C         | 0.17 c B            | 0.18 a A            | 0.16                |
|                        | F4 (15.90-15)                     | 0.15 b C         | 0.21 a A            | 0.17 b B            | 0.17                |
|                        | F5 (18.94-20)                     | 0.15 b B         | 0.18 b A            | 0.15 c B            | 0.16                |
| Mean of the N-total (%)|                                     | 0.16             | 0.18                | 0.17                | +                   |
| 125 days after transplanting | F0 (without fertilizer)         | 0.12 d B         | 0.14 d A            | 0.14 c A            | 0.15                |
|                        | F1 (20.69-0)                      | 0.14 b B         | 0.14 d B            | 0.16 c A            | 0.15                |
|                        | F2 (19.70-5)                      | 0.17 a A         | 0.16 c B            | 0.17 b A            | 0.16                |
|                        | F3 (17.27-10)                     | 0.13 c C         | 0.17 bc B           | 0.19 b A            | 0.16                |
|                        | F4 (15.90-15)                     | 0.14 b C         | 0.19 b B            | 0.20 a A            | 0.17                |
|                        | F5 (18.94-20)                     | 0.14 b B         | 0.20 a A            | 0.20 a A            | 0.16                |
| Mean of the N-total (%)|                                     | 0.15             | 0.17                | 0.16                | +                   |

Remarks: Values followed by the same lowercase letters in the same column are not significantly different according to Duncan’s multiple range test at the level of 5%. Values followed by the same uppercase letters in the same line are not significantly different according to Duncan’s multiple range test at the level of 5%.
The waterlogging level of 3 cm increased the pH \( \text{H}_2\text{O} \) compared to the waterlogging level of 0.5 cm (Table 4). Table 4 shows that the waterlogging level at 0 DAT and 125 DAT affects the value of pH \( \text{H}_2\text{O} \). Meanwhile, the composition of NP-SR fertilizer does not affect the value of pH \( \text{H}_2\text{O} \) in all observations. The increase in pH of the soil inundated is caused by reducing Fe\(^{3+}\) to Fe\(^{2+}\) when OH-liberation occurs and consumption of H\(^+\) (Bahmaniar & Mirnia, 2002). Dissociation of H\(^+\) from the edge of clay minerals and the surface of soil organic matter contributes to the soil’s acidity. Besides, acid-neutral-base salts in soil solutions derived from mineral weathering, decomposition of organic matter, and fertilization will affect the soil pH.

The pH \( \text{H}_2\text{O} \) had increased at 75 DAT in each waterlogging level treatment. The pH \( \text{H}_2\text{O} \) at 0 DAT classified as acidic in all waterlogging levels then as the plants grew, the pH \( \text{H}_2\text{O} \) became neutral. The changes in pH from acidic to neutral occurred from 0 DAT to 75 DAT. At each observation, the waterlogging level of 5 cm resulted in the lowest value of pH \( \text{H}_2\text{O} \). At a depth of 5 cm, pH \( \text{H}_2\text{O} \) also increased from 0 DAT to 75 DAT but then decreased at 125 DAT (6.2 to 6.0). The value of pH \( \text{H}_2\text{O} \) is strongly influenced by the level of waterlogging, which would determine ion solubility H\(^+\). In general, zeolite could increase soil pH because it has high alkalinity and contains alkaline cations, which can be released into the soil solution. Zeolite has a low base pH so that natural zeolite will neutralize the soil.

The value of pH KCl increased at 75 DAT in each of waterlogging level. A cation exchange causes the increase in pH KCl, in which K\(^+\) ions from KCl push H\(^+\) ions in the absorption complex, increasing H\(^+\) ions in the soil solution, then the H\(^+\) ion will be absorbed by zeolite so that the value of pH KCl will increase due to the decreasing H\(^+\) ion in the soil solution. KCl can measure the activity of H\(^+\) ions that are outside the soil solution. This is because K\(^+\) ions in KCl can be exchanged with H\(^+\) ions, while this does not apply to \( \text{H}_2\text{O} \) (Sutanto, 2005).

The value of soil EC ranges from 94.67 µS

| Treatment | pH H\(_2\)O | pH KCl | EC (µS.cm\(^{-1}\)) |
|-----------|------------|--------|---------------------|
|            | 0 DAT | 75 | 125 | 0 | 75 | 125 | 0 | 75 | 125 |
| Waterlogging level | | | | | | | | | |
| W0 | 5.51 b | 6.44 a | 6.50 a | 4.58 a | 5.61 a | 5.82 a | 108.11 a | 121.16 a | 136.72 a |
| W1 | 5.80 a | 6.44 a | 6.42 a | 4.45 b | 5.56 a | 5.78 a | 104.78 a | 113.66 a | 133.17 a |
| W2 | 5.51 b | 6.26 a | 6.12 b | 4.5 ab | 5.57 a | 5.84 a | 104.17 a | 111.11 a | 129.78 a |
| NP- SR fertilizer composition | | | | | | | | | |
| F0 | 5.54 a | 6.35 a | 6.23 a | 4.26 b | 5.56 a | 5.81 a | 114.78 a | 122.67 a | 147 a |
| F1 | 5.56 a | 6.37 a | 6.31 a | 4.55 b | 5.57 a | 5.88 a | 113.89 a | 119.89 a | 143.77 a |
| F2 | 5.69 a | 6.47 a | 6.45 a | 4.59 a | 5.62 a | 5.78 a | 99.00 a | 104.00 a | 113.78 b |
| F3 | 5.65 a | 6.43 a | 6.39 a | 4.56 a | 5.59 a | 5.75 a | 102.11 a | 108.56 a | 132.33 ab |
| F4 | 5.58 a | 6.37 a | 6.42 a | 4.55 a | 5.63 a | 5.83 a | 94.67 a | 96.89 a | 117.66 b |
| F5 | 5.6 a | 6.38 a | 6.12 a | 4.5 a | 5.52 a | 5.82 a | 109.67 a | 139.89 a | 144.77 a |
| CV (%) | 5.46 | 3.73 | 3.44 | 3.01 | 2.28 | 2.83 | 18.92 | 20.91 | 19.01 |

Remarks: Values followed by the same lowercase letters in the same column are not significantly different according to Duncan’s multiple range test at the level of 5%. W0 = 0.5 cm of water logging; W1 = 3 cm; W2 = 5 cm. F0 = Control without fertilizer application; F1 = NP-SR fertilizer with a composition of 20.69-0; F2 = 19.70-5; F3 = 17.27-10; F4 = 15.90-15; F5 = 18.94-20.
cm\(^{-1}\) to 114.78 µS cm\(^{-1}\) (0 DAT), 96.89 µS cm\(^{-1}\) to 139.89 µS cm\(^{-1}\) (75 DAT) and 117.66 µS cm\(^{-1}\) to 147 µS cm\(^{-1}\) (125 DAT). EC value tends to increase in line with the length of observation because during the planting period, natural zeolite from the soil sorption complex enters the soil to increase ionic strength and EC soil.

According to Table 5, the combination of NP-SR fertilizer with a composition of 15.90-15 (F4) and waterlogging level of 0.5 cm, 3 cm, and 5 cm increased rice grain density with the highest value compared to the other treatments at the same waterlogging level. The NP-SR fertilizer with a composition of 15.90-15 (F4) increased the percentage of rice grain to 82.63%. Meanwhile, the combination of NPSR fertilizer with a composition of 15.90-15 (F4) and waterlogging level of 3 cm (G1) increased the filled grain percentage up to 84.47%. The number of granules in each panicle depends on the nitrogen supply at the beginning of the panicle growth. Rice yields are significantly associated with adequate plant nitrogen levels at both critical growth stages (Sanchez, 1992).

The efficiency of N fertilizer in this study was calculated from the volatilization amount, quantifying how much N evaporated and utilized by the plant. The NPSR fertilizer composition and waterlogging level independently influence fertilizer efficiency. The waterlogging level of 3 cm and 5 cm can increase fertilizer efficiency (21.16 – 21.74%) compared to treatment without water. Meanwhile, in the treatment of waterlogging level of 0.5 cm, N fertilization efficiency was only 62.32%, and this might be due to the surface with shallow water, which quickly dries so that the condition is hotter than the other waterlogging.

NPSR fertilizer composition affects the efficiency of N fertilizer. Table 6 shows that the NPSR fertilizer composition F4 resulted in the highest efficiency of N fertilizer, reaching 75.99%, followed by F1, F2, F3, and F5, which showed N fertilizer efficiency of 73.94%, 72.94%, 72.66%and

| Table 5. Effects of the interaction between the application of NP-SR fertilizer composition and waterlogging level on the percentage of filled grain and empty grain |
|-----------------------------------------------|
| Observation | Composition of NP-SR fertilizer (K) | Waterlogging (W) | Mean (%) |
|-----------------------------------------------|
| 75 days after transplanting | F0 (without fertilizer) | W0 (0.5 cm) | 70.00 c A | 65.73 d B | 66.40 |
|  | F1 (20.69-0) | 71.17 bc B | 72.76 c B | 72.33 |
|  | F2 (19.70-5) | 71.61 bc C | 77.17 bc A | 75.11 |
|  | F3 (17.27-10) | 72.03 b C | 76.67 bc B | 75.63 |
|  | F4 (15.90-15) | 80.49 a C | 82.94 a B | 82.63 |
|  | F5 (18.94-20) | 73.06 b C | 79.45 b A | 76.20 |
| Mean (%) | 73.06 | 75.25 | 75.88 | + |
| 125 days after transplanting | F0 (without fertilizer) | W0 (0.5 cm) | 29.99 a C | 34.27 a B | 33.59 |
|  | F1 (20.69-0) | 28.83 b A | 27.24 b B | 27.66 |
|  | F2 (19.70-5) | 28.39 b A | 22.23 c C | 24.88 |
|  | F3 (17.27-10) | 27.97 bc A | 23.33 c B | 24.36 |
|  | F4 (15.90-15) | 19.51 c A | 17.06 d B | 17.36 |
|  | F5 (18.94-20) | 26.94 bc A | 20.55 dc C | 23.80 |
| Mean (%) | 26.94 | 24.78 | 24.11 | + |

Remarks: Values followed by the same lowercase letters in the same column are not significantly different according to Duncan’s multiple range test at the level of 5%. Values followed by the same uppercase letters in the same line are not significantly different according to Duncan’s multiple range test at the level of 5%. The density of rice grain is in %.
The results indicated that N loss through volatilization in the administration of NP-SR fertilizer composition ranged from 24.3% - 27.83% compared to urea treatment alone, reaching 70%. The same thing was revealed by Ahmed et al. (2010), mentioning that zeolite could minimize ammonia loss due to zeolites’ ability to absorb essential nutrients such as ammonium and potassium. Lija et al. (2012) reported that the soil treatment (control) did not show any loss of NH₃, and treatment with the addition of zeolite at a specific dose was significantly lower in the loss of NH₃ than in the treatment of urea without zeolite.

Omar et al. (2010) reported that zeolite mixtures significantly minimized volatile ammonia between 40 and 50% compared to urea without additives. Also, the treatment significantly increased available nitrogen nitrate compared to urea without additives. Volatilization of ammonia from urea could be significantly reduced in flooded conditions so that N efficiency increased.

During stagnant conditions, N loss occurs through evaporation, denitrification, and washing (Rahmawati 2013). Slowly released N fertilizers can reduce N loss and meet N needs in plants, as well as increase crop yields and fertilizer efficiency (Yi et al., 2006; Zhang et al., 2010; Gui-Hua et al., 2011; Chen et al., 2017). Slowly-released N fertilizers have a release speed that is almost equal to the N requirement absorbed by the plant, resulting in a reduction in inorganic N accumulation in the soil and minimizing the risk of N loss among N uptake by planting with N availability. N in the plant can be increased by providing balanced N, P, and K fertilizers (Bijay-Singh et al., 1995).

**CONCLUSION**

The level of waterlogging on a rice paddy field significantly affects the soil chemical properties, specifically the soil pH at the initial vegetative phase and harvest. The waterlogging level of 3-cm and the application of NP-SR fertilizer with a composition of 15.90-15 can reduce the volatilization rate, increase N fertilizer efficiency, and increase rice yield.

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**Table 6.** Evaporation of NH₃ and fertilizer efficiency as affected by the application of NP-SR fertilizer composition and waterlogging level

| Treatment | Fertilizer efficiency (%) |
|-----------|---------------------------|
| **Waterlogging** |                     |
| W0        | 62.32 b                  |
| W1        | 75.57 a                  |
| W2        | 75.87 a                  |
| **NP-SR fertilizer composition** |                     |
| F0        | 99.82 a                  |
| F1        | 73.00 bc                 |
| F2        | 72.95 c                  |
| F3        | 72.66 d                  |
| F4        | 73.95 b                  |
| F5        | 72.17 e                  |
| **CV (%)** | 4.62                     |

Remarks: Values followed by the same lowercase letters in the same column are not significantly different according to Duncan’s multiple range test at the level of 5%. W0 = 0.5 cm of waterlogging; W1 = 3 cm; W2 = 5 cm. F0 = Control without fertilizer application; F1 = NP-SR fertilizer with a composition of 20.69-0; F2 = 19.70-5; F3 = 17.27-10; F4 = 15.90-15; F5 = 18.94-20.
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