One-pot, sequential four-component synthesis of novel heterocyclic [3.3.3] propellane derivatives at room temperature†

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An efficient, one-pot, two-step, four-component reaction for the synthesis of propellane derivatives is described by the condensation reaction betweenacenaphthenequinone, malono derivatives, primary amines and β-ketoester or β-diketone derivatives in the presence of triethylamine in ethanol at room temperature. Using this procedure, all the products were obtained in good to excellent yields.

Introduction

In modern organic chemistry, the improvement of reaction efficiency, the reduction of waste, the avoidance of toxic reagents and the responsible utilization of our resources have become critical objectives.1,2 Multi-component reactions, are useful tools for the synthesis of biologically and pharmacologically important compounds because of their environmentally friendly, green characteristics and atom efficient.3–6 Therefore, the design and application of multi-component reactions has attracted great attention from research groups working in areas such as biological, medicinal and organic chemistry.

Heterocyclic compounds occur commonly in nature and are essential to life.7 These compounds have received considerable attention in recent times due to their wide applications. Functionalized nitrogen and oxygen-containing heterocycles are of importance as they have various industrial, pharmaceutical and biological applications.8–20

Propellanes are tricyclic compounds in which the three rings are connected to a common C–C single bond.21 Nitrogen- and oxygen-containing propellanes and their analogues constitute the structural framework of many natural and non-natural products such as periglaucine A, hasubanone, sinoacutine and merrilactone A (Fig. 1).22,23 Propellanes due to their challenging framework and broad spectra of biological and pharmacological activities, constitute an interesting class of compounds in organic chemistry.24

Recently, there have been some reports about the synthesis of propellane compounds from ninhydrin25–32 and acenaphthoquinone.33–34 In continuation of our research on the synthesis of biologically important heterocycles and green chemistry protocols,35–47 we report a simple and efficient method for the synthesis of novel propellane derivatives via a one-pot, sequential four-component condensation reaction between acenaphthoquinone (1), malono derivatives (2), primary amines (4) and β-ketoester or β-diketone (5) derivatives in the presence of triethylamine in ethanol at room temperature (Scheme 1).

Results and discussion

In the first step, the reaction between acenaphthoquinone (1), malononitrile (2a), ethyl acetoacetate (5a) and benzyl amine (4c), as a model reaction, was investigated in the presence of different molar ratios of Et3N in some solvents (5 mL) at room temperature and the results are summarized in Table 1. For this propose to a stirred mixture of acenaphthoquinone (1 mmol) and malononitrile 2a (1 mmol) in EtOH (5 mL) at room temperature different amount of Et3N were added to form the corresponding Knoevenagel adduct and then ethyl acetoacetate 5a (1 mmol) and benzyl amine 4c (1 mmol) were added to obtain the corresponding product 6a. As it is shown in Table 1, higher yield and shorter reaction time were obtained when the reaction was carried out in the presence of 1 mmol of the Et3N in ethanol at room temperature (Table 1, entry 4). It should be mention that simultaneity of this four-component condensation reaction was studied in the presence of 1 mmol Et3N in EtOH at room temperature and we observed that benzyl amine has remained

Fig. 1 Typical natural and biological products containing propellane skeleton.
intact in this condition and the corresponding propellane was not formed.

In the next step, the scope and efficiency of the process was explored under the optimized conditions. For this purpose, acenaphthoquinone was condensed with alkyl malonates, aliphatic and aromatic amines and dicarbonyl compounds in the presence of Et₃N (1 mmol) to afford the corresponding products. The structural diversity of reactants is summarized in Fig. 2 and the results are displayed in Table 2.

The synthetic pathway for the synthesis of titled compounds is consisting of two steps. At first, compound 3 are obtained from acenaphthoquinone 1 and alkyl malonates 2. Then, the resulting products are treated with primary amines 4 and β-dicarbonyl compounds 5 to afford the related propellane derivatives as the desired products. As Table 2 indicates, a variety of alkyl malonates, aliphatic and aromatic amines and β-dicarbonyl compounds were successfully applied in this process to afford the corresponding propellane derivatives as novel compounds in good to excellent yields.

To demonstrate the generality and scope of this method, malono derivatives, various acyclic β-dicarbonyl compounds such as methyl acetoacetate, ethyl acetoacetate and ethyl benzoylacacetate were treated with a broad range of aliphatic and aromatic amines in the presence of Et₃N (1 mmol) in EtOH at room temperature and the results are shown in Table 2. Use of malononitrile leads to shorter reaction times and higher yields compared to application of other alkylmalonates such as methyl cyanoacetate or ethyl cyanoacetate (compare 6b with 6p and 6l with 6q). As it is shown in Table 2, benzyl amines bearing electron-donating or -withdrawing substituents gave the desired propellanes in high yields and in short reaction times and it seems that when the methoxy group is placed on the aromatic ring, reactivity is better than when the chlorine group is on the ring. Moreover, the presence of 3,5-dimethyl group on the aromatic ring has the best efficiency. Some aromatic amines were also screened to carry out the four-component reaction by this method and the results are listed in Table 2 (entries 6h–6j). Generally, aliphatic amines are more reactive than aromatic amines. Ortho-substituted anilines, whatever the nature of the substituted groups, required a longer reaction period. For instance, reaction of acenaphthenoquinone, malononitrile and 2,6-dimethylaniline with ethyl acetoacetate gave the desired product (6j) in 70% yield after 3.1 h. Moreover, aliphatic amines such as ethyl amine and butyl amine were used for the synthesis of the corresponding product in excellent yields (Table 2, entries 6l–6o). In the same way, Some cyclic and acyclic 1.3-dicarbonyl compounds were also screened to carry out the one-pot four-component reaction by this method and the results are listed in Table 2. As it is shown in Table 2, aliphatic β-keto esters are more reactive than aromatic β-keto esters (compare 6a and 6c) and no products were obtained when cyclic 1.3-dicarbonyl compounds were involved in this one-pot room temperature reaction.

| Entry | Amount of Et₃N (mmol) | Solvent | Time (h) | Yieldb (%) |
|-------|----------------------|---------|----------|------------|
| 1     | —                    | EtOH    | 5        | —          |
| 2     | 0.1                  | EtOH    | 5        | 45         |
| 3     | 0.5                  | EtOH    | 5        | 87         |
| 4     | 1.0                  | EtOH    | 1.2      | 91         |
| 5     | 1.5                  | EtOH    | 1.2      | 91         |
| 6     | 1.0                  | H₂O     | 2.0      | 70         |
| 7     | 1.0                  | DMF     | 4.0      | 60         |
| 8     | 1.0                  | MeOH    | 1.5      | 85         |
| 9     | 1.0                  | CHCl₃   | 4.0      | 40         |
| 10    | 1.0                  | THF     | 4.0      | 55         |
| 11    | 1.0                  | CH₃CN   | 40       | 45         |

*a* Triethylamine (1.0 mmol) was added to a stirred mixture ofacenaphthoquinone 1 (1 mmol), malononitrile 2a (1 mmol) in EtOH (5 mL) at room temperature and after about 10 min to form the corresponding Knoevenagel adduct, then ethyl acetoacetate 5a (1 mmol) and benzyl amine 4c (1 mmol) were added to obtain the corresponding product 6a. "Isolated yields.

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**Fig. 2** Diversity elements employed for synthesis of propellane derivatives.
The proposed mechanism for the synthesis of propellane compounds in the presence of Et$_3$N is shown in Scheme 2. Based on this mechanism, Et$_3$Ni is effective catalyst for the formation of olefin (3) which readily prepares in situ by Knoevenagel condensation of acenaphthoquinone (1) with the active methylene of compound (2). Next, the formation of enamine (7) occurs through condensation of amine (4) with dicarbonyl compound (5). Then, the enamine (7) attacks to olefin (3) in a Michael addition to produce intermediate (8). After proton transfer and tautomization, intermediate (8) converts to intermediate (10), through intramolecular cyclization by nitrogen attack of enamine to carbonyl group. In this stage, nucleophilic addition of amino group to C=O bond afforded intermediate (10). Then, intramolecular O-cyclization and the tautomerization of imino group to amino group lead to propellanes (6).

**Experimental**

All chemicals were purchased from Merck or Fluka chemical companies. The $^1$H NMR (400 MHz) and $^{13}$C NMR (100 MHz) were run on a Bruker Avance 400. Melting points were recorded on a Stuart Scientific Apparatus SMP3 (UK) in open capillary tubes. Elemental C, H and N analyses were performed using a Costech CHNS-O elemental analyzer.

**General procedure for the synthesis of propellane derivatives 6**

Triethylamine (1.0 mmol) was added to a stirred mixture of acenaphthoquinone 1 (1 mmol), malono derivatives 2 (1 mmol) in EtOH (5 mL) at room temperature to form the corresponding Knoevenagel adduct (about 10–30 min), then dicarbonyl compound 5 (1 mmol) and primary amine 4 (1 mmol) were added to obtain the desired product 6. Isolated yields.

**Table 2** One-pot, sequential four component synthesis of propellane derivatives in the presence of Et$_3$N (1 mmol) in EtOH at room temperature.$^{a,b}$

| 6a, 1.2 h, 91% | 6b, 1.2 h, 92% | 6c, 2.0 h, 87% |
| 6d, 1.6 h, 90% | 6e, 1.2 h, 92% | 6f, 1.5 h, 86% |
| 6g, 1.0 h, 95% | 6h, 2.2 h, 79% | 6i, 2.0 h, 83% |
| 6j, 3.1 h, 70% | 6k, 2.7 h, 77% | 6l, 1.2 h, 91% |
| 6m, 1.2 h, 93% | 6n, 1.5 h, 87% | 6o, 1.3 h, 89% |
| 6p, 2.0 h, 86% | 6q, 2.2 h, 85% |

$^a$Triethylamine (1.0 mmol) was added to a stirred mixture of acenaphthoquinone 1 (1 mmol), malono derivatives 2 (1 mmol) in EtOH (5 mL) at room temperature to form the corresponding Knoevenagel adduct (about 10–30 min), then dicarbonyl compound 5 (1 mmol) and primary amine 4 (1 mmol) were added to obtain the desired product 6. Isolated yields.

**Ethyl-11-amino-7-benzyl-10-cyano-8-methyl-7H-6b,9a-(epoxypetheno)acenaphtho[1,2-b]pyrrole-9-carboxylate 6a.** White powder, mp = 249–251 °C, $^1$H NMR (DMSO-$d_6$, 400 MHz) δ (ppm): 1.45 (t, J = 6.6 Hz, 3H), 2.25 (s, 3H), 4.04 (q, J = 7.2 Hz, 2H), 4.54 (d, J = 16.0 Hz, 1H), 4.60 (d, J = 16.0 Hz, 1H), 7.57 (s, 2H), 7.66–7.71 (m, 3H), 7.73–7.81 (m, 4H), 7.89 (d, J = 8.0 Hz, 2H), 8.04 (d, J = 7.2 Hz, 2H). $^{13}$C NMR (DMSO-$d_6$, 100 MHz)
δ (ppm): 14.4, 14.6, 58.1, 60.5, 61.2, 70.7, 79.9, 110.4, 118.4, 120.6, 121.0, 124.7, 125.8, 127.8, 129.2, 129.6, 132.0, 135.8, 136.3, 142.7, 147.2, 147.6, 147.7, 162.4, 166.5, 166.9, 168.0, 168.1. Anal. calcd for C28H23N3O3: C, 74.82; H, 5.16; N, 9.35%. Found: C, 74.85; H, 5.14; N, 9.33%. IR (KBr) νmax cm⁻¹: 3358 and 3192 (NH2), 1931 (CN), 1687 (C=O), 1652 (O=C).

### 9-Acetyl-11-amino-7-benzyl-8-methyl-7H-6b,9a-(epoxyetheno)acenaphthof[1,2-b]pyrrole-9-carbonylitrile 6b.
White powder, mp = 250–252 °C, 1H NMR (DMSO-δ6, 400 MHz) δ (ppm): 2.32 (s, 3H), 2.36 (s, 3H), 3.45–3.49 (m, 1H), 3.79–3.82 (m, 1H), 4.77 (d, J = 16.0 Hz, 1H), 5.03 (d, J = 15.6 Hz, 1H), 7.57 (s, 2H), 7.66–7.71 (m, 2H), 7.73–7.77 (m, 3H), 7.85 (d, J = 7.6 Hz, 2H), 8.02 (d, J = 8.0 Hz, 2H).

### 9-Acetyl-11-amino-7-benzyl-8-cyano-7H-6b,9a-(epoxyetheno)acenaphthof[1,2-b]pyrrole-9-carbonitrile 6c.
White powder, mp = 250–253 °C, 1H NMR (DMSO-δ6, 400 MHz) δ (ppm): 2.05 (s, 3H), 2.10 (s, 3H), 2.31 (s, 3H), 2.36 (s, 3H), 4.96 (d, J = 6.0 Hz, 1H), 5.02 (d, J = 16.0 Hz, 1H), 6.83 (d, J = 7.6 Hz, 1H), 7.02 (t, J = 7.4 Hz, 1H), 7.21–7.35 (m, 2H), 7.49 (d, J = 7.2 Hz, 2H), 7.58 (d, J = 9.2 Hz, 1H), 7.79 (s, 1H), 7.85 (dd, J = 2.4 Hz, 8.8 Hz, 1H), 8.02 (d, J = 2.4 Hz, 1H).

### 9-Acetyl-11-amino-7-(3,5-dimethylbenzyl)-8-methyl-7H-6b,9a-(epoxyetheno)acenaphthof[1,2-b]pyrrole-10-carbonitrile 6g.
White powder, mp = 250–253 °C, 1H NMR (DMSO-δ6, 400 MHz) δ (ppm): 1.73 (s, 3H), 2.36 (s, 3H), 2.66 (s, 2H), 3.70–3.77 (m, 4H), 3.8 (d, J = 6.8 Hz, 1H), 3.87 (d, J = 7.2 Hz, 1H), 7.9 (d, J = 8.0 Hz, 2H), 8.07 (d, J = 8.0 Hz, 2H).

### 9-Acetyl-11-amino-7-(4-methoxyphenyl)-8-methyl-7H-6b,9a-(epoxyetheno)acenaphthof[1,2-b]pyrrole-10-carbonitrile 6h.
White powder, mp = 247–249 °C, 1H NMR (DMSO-δ6, 400 MHz) δ (ppm): 1.38 (t, J = 7.0 Hz, 3H), q, J = 6.8 Hz, 2H), 4.92 (d, J = 16.0 Hz, 1H), 4.98 (q, J = 16.0 Hz, 1H), 7.43 (t, J = 7.4 Hz, 3H), 7.51 (t, J = 7.0 Hz, 3H), 7.66 (s, 2H), 7.70–7.77 (m, 4H), 7.83 (d, J = 6.8 Hz, 1H), 7.87 (d, J = 7.2 Hz, 1H), 7.93 (d, J = 8.0 Hz, 2H), 8.07 (d, J = 8.0 Hz, 2H).

### 9-Acetyl-11-amino-7-(4-methoxystyryl)-8-methyl-7H-6b,9a-(epoxyetheno)acenaphthof[1,2-b]pyrrole-10-carbonitrile 6i.
White powder, mp = 247–249 °C, 1H NMR (DMSO-δ6, 400 MHz) δ (ppm): 1.38 (t, J = 7.0 Hz, 3H), 2.37 (s, 3H), 3.75 (s, 3H), 4.67 (d, J = 16.0 Hz, 1H), 4.78 (d, J = 16.0 Hz, 1H), 7.35 (d, J = 6.8 Hz, 2H), 7.55–7.59 (m, 2H), 7.77 (t, J = 7.6 Hz, 2H), 7.84–7.89 (m, 2H), 8.09 (s, 2H), 8.17 (d, J = 7.6 Hz, 2H).

### 9-Acetyl-11-amino-7-(4-chlorobenzyl)-8-methyl-7H-6b,9a-(epoxyetheno)acenaphthof[1,2-b]pyrrole-9-carboxylic acid 6j.
White powder, mp = 249–251 °C, 1H NMR (DMSO-δ6, 400 MHz) δ (ppm): 1.44 (t, J = 7.0 Hz, 3H), 2.25 (s, 3H), 4.11–4.17 (m, 2H), 4.42 (d, J = 16.0 Hz, 1H), 4.48 (d, J = 16.0 Hz, 1H), 7.77 (s, 2H), 7.85–7.91 (m, 2H), 7.92–7.96 (m, 3H), 7.99 (t, J = 6.0 Hz, 1H), 8.08 (d, J = 8.4 Hz, 2H), 8.23 (d, J = 8.0 Hz, 2H).

### 9-Acetyl-11-amino-7-(2,6-dimethylphenyl)-8-methyl-7H-6b,9a-(epoxyetheno)acenaphthof[1,2-b]pyrrole-9-carboxylic acid 6k.
White powder, mp = 255–257 °C, 1H NMR (DMSO-δ6, 400 MHz) δ (ppm): 1.35 (t, J = 7.2 Hz, 3H), 2.21 (s, 3H), 2.30 (s, 3H), 3.31–3.34 (m, 2H), 3.75 (d, J = 6.8 Hz, 2H), 7.55–7.59 (m, 2H), 7.77 (t, J = 7.6 Hz, 2H), 7.84–7.89 (m, 2H), 8.10 (s, 2H), 8.17 (d, J = 7.6 Hz, 1H).

### 9-Acetyl-11-amino-7-(2,6-dimethylphenyl)-8-methyl-7H-6b,9a-(epoxyetheno)acenaphthof[1,2-b]pyrrole-9-carboxylic acid 6l.
White powder, mp = 249–251 °C, 1H NMR (DMSO-δ6, 400 MHz) δ (ppm): 1.44 (t, J = 7.0 Hz, 3H), 2.25 (s, 3H), 4.11–4.17 (m, 2H), 4.42 (d, J = 16.0 Hz, 1H), 4.48 (d, J = 16.0 Hz, 1H), 7.77 (s, 2H), 7.85–7.91 (m, 2H), 7.92–7.96 (m, 3H), 7.99 (t, J = 6.0 Hz, 1H), 8.08 (d, J = 8.4 Hz, 2H), 8.23 (d, J = 8.0 Hz, 2H).
Ethyl-11-amino-7-(4-chlorophenyl)-10-cyano-8-methyl-7H-6b,9a-(epoxetheno)acenaphtho[1,2-b]pyrrole-9-carboxylate 6k. White powder, mp = 251–253 °C, 1H NMR (DMSO-d6, 400 MHz) δ (ppm): 1.19 (t, J = 7.2 Hz, 3H), 2.00 (s, 3H), 4.04 (q, J = 7.2 Hz, 2H), 2.56 (d, J = 8.8 Hz, 1H), 2.75 (d, J = 7.2 Hz, 1H), 2.77 (t, J = 7.6 Hz, 1H), 2.79 (s, 2H), 2.84 (dd, J = 2.4 Hz, 8.8 Hz, 1H), 7.00 (t, J = 7.6 Hz, 1H), 7.08–8.06 (m, 3H), 8.37 (d, J = 8 Hz, 2H). 13C NMR (DMSO-d6, 100 MHz) δ (ppm): 13.7, 14.6, 59.8, 60.2, 71.9, 79.2, 103.0, 108.9, 113.1, 116.9, 119.8, 123.5, 125.3, 126.1, 134.1, 136.2, 142.4, 151.2, 152.4, 155.1, 158.7, 159.3, 167.2, 167.7, 180.2. Anal. calc'd for C24H23N3O2: C, 74.01; H, 4.29; Cl, 7.54; N, 8.94%. Found: C, 69.05; H, 4.26; N, 8.92%.

4-Acetyl-11-amino-7-ethyl-8-methyl-7H-6b,9a-(epoxetheno)acenaphtho[1,2-b]pyrrole-9-carboxylate 6m. White powder, mp = 258–260 °C, 1H NMR (DMSO-d6, 400 MHz) δ (ppm): 0.76 (t, J = 7.2 Hz, 3H), 1.44 (t, J = 7.0 Hz, 3H), 2.25 (s, 3H), 2.81 (q, J = 7.0 Hz, 2H), 4.30 (q, J = 7.0 Hz, 2H), 7.57 (s, 2H), 7.66–7.71 (m, 1H), 7.72–7.76 (m, 2H), 7.79 (t, J = 6.0 Hz, 1H), 7.88 (d, J = 8.4 Hz, 1H), 8.03 (d, J = 8.0 Hz, 1H). 13C NMR (DMSO-d6, 100 MHz) δ (ppm): 7.9, 14.4, 14.6, 58.1, 58.6, 60.5, 70.7, 79.9, 110.4, 118.4, 120.6, 121.0, 124.7, 125.8, 128.7, 129.1, 129.6, 132.0, 135.8, 136.3, 142.7, 167.3, 168.1. Anal. calc'd for C24H23N3O2: C, 71.30; H, 5.46; N, 10.85%. Found: C, 71.28; H, 5.44; N, 10.88%. IR (KBr) (v (cm−1)): 3405 and 3278 (NH3), 2197 (CN), 1663 (C=O), 1639 (O=C=O).

References
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Conclusions
In conclusion, we have reported a highly efficient method for the synthesis of biologically important propellane derivatives via a one-pot, sequential four-component condensation reaction presence of triethylamine in ethanol at room temperature. High atom-economy, simple procedure in the excellent yields, easy workup procedure, mild reaction conditions and metal-catalyst free are the main advantages of this method.

Conflicts of interest
There are no conflicts to declare.

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