Ultrahigh Radio Frequency (820-960 MHz) Identification System Affects Antioxidant Barrier in Red Blood Cell Concentrates

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Abstract

Background and objectives: Radio frequency identification (RFID) tags have advantages over bar codes in blood management system. Ultra-high frequency (UHF) RFID technology (820-960 MHz) in many ways has more advantages than high frequency (HF) RFID (13.56 MHz). The aim of our study was the evaluation of the effects of UHF RFID in comparison to HF RFID radiation on quality markers of RBC and on the oxidoreductive balance in RBC stored in containers labelled with RFID tags.

Materials and methods: Ten RBC units were split into three components, one control unit-marked with bar code and two test units labeled with RFID tags - one operating on UHF and the second on HF radio waves. All units were stored at 2-6°C for 35 days. The test groups were exposed to radio waves continuously during storage. The quality parameters as well as concentration of malondialdehyde (MDA), and activities of superoxide dismutase, glutathione peroxidase and glutathione reductase were measured.

Results: The degree of hemolysis and the concentration of K+ were statistically significant higher on 35th day of storage in RBC labelled with UHF RFID tags compare to control and HF RFID groups. On the 35th day of storage the concentration of MDA was statistically significant higher and the activities of oxidoreductive enzymes were statistically significant lower in the UHF group compare to control and HF groups.

Conclusion: UHF RFID tags affect oxidoreductive balance in RBC and may lead to eryptosis. The approval of UHF RFID system for blood components needs further studies.

Keywords: RFID; Radio waves; Erythrocytes; Antioxidant barrier; Tag

Introduction

In recent years, there is growing interest in radio frequency identification (RFID) technology as a complement to bar code technology in transfusion medicine. Several blood centers and hospitals have tested the utilization of radio frequency identification (RFID) as a means of enhancing tracking, monitoring, labeling, and storage in the blood transfusion supply chain, and have found it to be very promising [1]. With the help of built-in memory, RFID tag can carry defined information on the blood component been tagged. RFID can read multiple numbers of tags simultaneously even in adverse light conditions [2]. By exploiting these characteristics, RFID can be employed in healthcare especially in transfusion medicine.

RFID system is composed of transponder tags, readers, and a hardware system to which information is written. There are three different frequency ranges on which RFID can work. Low frequency (LF) bands typically work at frequencies of 125-134 kHz. High Frequency (HF) bands operate at 13.56 MHz. This is the global ISO-standardized frequency, and is widely accepted for use in the healthcare industry. Ultra High Frequency (UHF) works at 850-900 MHz. These are the most expensive tags, possessing the best ranges and transferring data at the fastest. UHF RFID tags are better than HF RFID tags in either in the read and write times, distance from tag to reader, and the number of targets that can be read simultaneously or on the environmental adaptability and anti-interference ability. Particularly UHF RFID tags are not affected by ultralow temperature. It makes them more suitable for applications in the management of blood components.

However, UHF licenses vary in allowance due to health and safety issues. The International Society for Blood Transfusion (ISBT) working party on RFID has recommended HF (13.56 MHz) as the international standard for transfusion medicine but the standards for UHF RFID can vary from country to country [2,3]. RFID technology is used mainly for identification of valuable commodities such as blood components albeit it can support additional functionalities like monitoring temperature. Even though the effects of radio frequency (RF) electromagnetic radiation on biologics due to temperature changes are relatively well understood, there are few studies in the literature about other effects of this kind of radiation that can occur without a noticeable change in temperature.

There is little known about the effect of radio waves on oxidative stress and there are no data on the RFID effect on oxidant-antioxidant equilibrium in blood components. The aim of our study was the evaluation of the effects of 850-900 MHz RF radiation in comparison to 13.56 MHz RF radiation on quality markers of RBC and on the intensity of lipid peroxidation and activity of antioxidant barrier enzymes in RBC stored in containers labelled with RFID tags.

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Materials and Methods

All experiments were approved by the University Bioethics Committee of the Medical University of Białystok (R-1-002/445/2012). Blood donations were conducted in accordance with the Public Blood Transfusion Service Act sanctioned in 1997. Written informed consent for this study was obtained from each blood donor.

Preparation of RBCs

Ten units of whole blood (450 ± 50 mL) were collected into standard triple containers 450/2×600 RW (Fresenius AG, Schweinfurt, Germany) with 63 ml CPD (Citrate, Phosphate, Dextrose) as an anticoagulant. Before separation of blood components, the units of blood were pre-stored for 2.5 hours at 22 ± 2°C. Then whole blood units were centrifuged using Roto Silenta 630RS (Hettich, Tuttinglen, Germany) with the acceleration of 4946 g (3926 rpm/min) for 10 minutes at 22°C. Separation of whole blood plasma anduffy coat (BC) was made by using the automatic press Compomat G4 (Fresenius, Schweinfurt, Germany). Erythrocytes were suspended in SAGM (Saline, Adenine, Glucose, Mannitol) preservative solution (100 mL) which allows their storage for 42 days at 2-6°C.

RFID System

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The readers have been continuously reading tags in millisecond intervals. All components were stored at 2-6°C for 35 days in refrigerators of the same manufacturer and model which were placed in separate rooms in order to avoid intergroup interference.

Sample collection

Before sampling RBCs were gently mixed and samples were taken into two test-tubes (about 10 mL in each test tube) (without anticoagulant) for biochemical analysis and assessment of oxydoreductive balance. Samples were taken on day 0 immediately after preparation of RBC, and on day 7, 14, 21 and 35 of storage. The samples were centrifuged at 3000xg for 10 min. After aspiration of plasma the erythrocytes were washed with 0.9% NaCl solution. Erythrocytes were hemolysed by adding triple volume of distilled water. The Hb concentration in the hemolysate was measured by Pentra 80 ABX (Horiba Diagnostic, France) using the Drabkin method [4-8].

Determination of Lipid Peroxidation and Antioxidant Enzyme Activity

Lipid peroxidation

As a marker of lipid peroxidation malonaldehyde (MDA) concentration was determined in isolated erythrocytes using the method described by Buege and Aust [9]. This method is based on reaction with thio-barbituric acid (TBA) and is used for the production of a color MDA-TBA complex in acidic conditions at an elevated temperature. Erythrocytes were incubated in the presence of 20% TCA at 4°C for 1 h, and then centrifuged at 1000 x g for 5 min. The obtained supernatant in the presence of 0.26M TBA was heated for 15 min at 100°C. Absorbance measurement was performed at λ=532 nm (Spectrophotometer Epoll-20, Poll Ltd., Poland) and was compared with first control sample (20% TCA, 0.26 M TBA) and second control sample (20% TCA and erythrocytes). Amplification of peroxidation during the assay was stopped by the addition of the chain-breaking antioxidant butylated hydroxytoluene (10 µL of 2 mM/L per 1 mL of RBC). The concentration of MDA in a sample was calculated using a molar extinction coefficient 1.56 × 10⁶ mol⁻¹·cm⁻¹ and expressed in nanomoles per gram of hemoglobin [nmol/gHb].

Superoxide dismutase activity (SOD)

Activity of SOD was determined using the method of Sykes et al. [10]. Previously prepared hemolysates were used to measure the activity of SOD in the tested samples. Three and half ml of bidistilled water chilled to +4°C were added to 0.5 ml of the hemolysate, followed by 1 ml of 96% ethanol and 0.6 ml of chloroform. The mixture was shaken and then centrifuged for 10 minutes at 4200 × g. The aqueous supernatant containing the enzyme was used in the respective reaction mixture. The reaction mixture contained 1.8 mL carbonate buffer (0.05 mol/L), 100 µL of supernatant and 100 µL of adrenaline (18 mg/10 mL). The increase in absorbance at 490 nm (Spectrophotometer Epoll-20, Poll Ltd., Poland) (T=30°C) from zero during 2 min was compared
with control in which the red blood cells were replaced by 0.9 % NaCl. The method involves the enzyme-induced inhibition of adrenalin self-
oxidation to adrenochrome in acidic conditions accompanied by the generation of superoxide radical (O2·-) as an intermediate product. The activity of SOD was measured in one-minute time intervals based on changes in absorbance (λ=490 nm, T=30°C) in a sample containing the enzyme as compared to the analogous time absorbance changes in control samples. The activity of SOD in study samples was expressed in adrenaline units [U/gHb].

Glutathione peroxidase activity (GSH-Px)

The activity of GSH-Px was determined using the method described by Paglia and Valentine [11]. The reaction mixture contained 100 µL of NADPH solution (8.4 mM/L), 10 µl glutathione reductase (GSSG-R) (100 U/mL), 10 µL sodium azide (1.125 mol/l), 100 µL glutathione (0.02 mol/L) filled with phosphate buffer (0.05 mol/L) up to the volume of 2.8 mL and 100 µL hemolysate. The enzymatic reaction was initiated by addition of 100 µL of 0.022 mol/l H2O2. In the presence of GSSG-R and NADPH the oxidized glutathione (GSSG) is immediately converted to the reduced form with a concomitant oxidation of NADPH to NADP+. The samples were assayed by UV Visible spectrophotometer (Epoll-20, Poll Ltd., Poland) at a wavelength of 340 nm. The activity of GSH-Px was expressed in international units [U/gHb].

Glutathione reductase activity

The activity of GSSH-R was determined using the method described by Mize and Langdon [12]. The reaction mixture contained 0.5 mL potassium phosphate solution (pH 7; 0.02 M), 0.5 mL EDTA (0.33 mM), 0.5 mL GSSH (7 mM/L) and 200 µL of NADPH solution (8.4 mM/L) and 0.04 mL hemolysate. The rate of change of extinction at 340 nm was determined by UV Visible spectrophotometer (Epoll-20, Poll Ltd., Poland) for one minute at 37°C and was compared with control (Tris-HCl buffer). The activity of GSSH-R was expressed in international units [U/gHb].

Hemogram

Complete hemogram: hemoglobin, hematocrit, total count of WBC, total count of RBC, platelet count, mean corpuscular volume, mean corpuscular hemoglobin, and mean corpuscular hemoglobin concentration were estimated by Pentra 80ABX (Horiba Diagnostic, France) automated cell counter. Concentration of free hemoglobin was measured using Plasma Low HB System (Hemocure, Sweden) according to the manufacturer’s recommendation and the degree of hemolysis—the percent of free hemoglobin in relation to the total was calculated using the formula: (100-Hct) × plasma Hb(g/dL)/total Hb(g/dL), where Hct is hematocrit and Hb is hemoglobin concentration.

Electrolytes, pH and lactate dehydrogenase

The concentration of potassium, sodium and chloride ions were estimated by Pentra 80ABX (Horiba Diagnostic, France) automated cell counter. Concentration of free hemoglobin was measured using Plasma Low HB System (Hemocure, Sweden) according to the manufacturer’s recommendation and the degree of hemolysis—the percent of free hemoglobin in relation to the total was calculated using the formula: (100-Hct) × plasma Hb(g/dL)/total Hb(g/dL), where Hct is hematocrit and Hb is hemoglobin concentration.

Statistic

In the statistical analysis, there was no normal distribution of quantitative variables analyzed, which was checked using Shapiro-Wilk test. Comparing the quantitative variables without the normal distribution was used nonparametric ANOVA Kruskal-Wallis Rank with the test post-hoc multiple comparisons average rank for all samples, for the three groups. Results were considered statistically significant at p<0.05. Elements of descriptive statistics were used to determine medians and quartiles for each parameter. Statistica 10.0 (StatSoft, Krakow, Poland) was used for statistical analyses.

Results

The degree of hemolysis in the test and control RBC units has increased with time of storage. On day 35 in RBC units from group UHF hemolysis degree was statistically significant higher (p<0.05) than in other tested RBC groups. In each study groups hemolysis degree was within the European Council criteria (lower than 0.8%) (Table 1).

Biochemical parameters

The concentration of glucose and pH value has decreased during the storage period in RBC units but they were not statistically significant different between the groups. The activity of LDH has increased in the units of RBC during storage (Table 1). There was no statistically significant differences in the activity of LDH between treated groups. The concentration of K+ has increased during storage in each study groups. On day 35 in RBC units from group UHF statistically significant higher (p<0.05) concentration of K+ was noted in comparison to other tested groups. The concentration of Na+ ions did not show statistically significant differences between the groups of RBC units during storage (Table 1).

Lipid peroxidation

The concentration of MDA in RBC units have increased with the time of storage. There were no statistically significant differences in the concentration of MDA between the groups until day 35. The concentration of MDA in the group UHF subjected to constant exposure to UHF radio waves on the 35th day of storage was statistically significant higher than in control group and group HF (13.5 MHz) (Figure 1A).

Antioxidative status biomarkers

The activity of SOD in all groups has decreased with time during the 35-day storage. There were no statistically significant differences in the activity of SOD between the groups until day 35. The activity of SOD on day 35 in group UHF was statistically significant lower than using the Konelab 60ISE apparatus (BioMerieux, France). pH value was determined at 22°C with a pH-meter Suntex SP-701 (Poland). The activity of LDH was assessed using Cobas c311 System (Roche, Switzerland).

| Time (day) | pH | LDH (U/I) | Hemolysis rate (%) | K+ (mM/mL) | Na+ (mM/mL) |
|-----------|----|-----------|--------------------|------------|-------------|
| Control   | Group HF | Group UHF | Control | Group HF | Group UHF | Control | Group HF | Group UHF | Control | Group HF | Group UHF |
| 0         | 7.49  | 7.49     | 7.49    | 80.3 | 80.3 | 80.3 | 0.11 | 0.11 | 0.11 | 2.44 | 2.44 | 2.44 | 138.4 | 138.4 | 138.4 |
| 7         | 7.27  | 7.27     | 7.27    | 137.5 | 140.2 | 139.2 | 0.29 | 0.29 | 0.29 | 20.45 | 20.9 | 21.0 | 132.4 | 131.9 | 131.7 |
| 14        | 7.04  | 7.00     | 7.04    | 200.4 | 210.4 | 218.8 | 0.38 | 0.38 | 0.4 | 29.78 | 28.6 | 29.2 | 126.05 | 127.2 | 126 |
| 21        | 6.92  | 6.92     | 6.88    | 263.7 | 275.8 | 277.8 | 0.5 | 0.48 | 0.52 | 35.6 | 34.6 | 36.2 | 121.0 | 122.4 | 120.8 |
| 35        | 6.68  | 6.66     | 6.61    | 300.2 | 299.8 | 308.2 | 0.6 | 0.58 | 0.68* | 39.7 | 40.2 | 41.8* | 110.30 | 113.3 | 110.2 |

Table 1: Effect of RFID on hemolysis rate, pH, LDH activity, K+, Na+ concentrations in stored units of RBC, n=10 (*p<0.05).
control and group HF (p<0.05) (Figure 1B). The activity of GSH-Px has decreased in all groups of RBC units during storage. There were no statistically significant differences in the activity of GSH-Px between the groups until day 35. The activity of GSH-Px in group UHF on day 35 was statistically significant lower than in other groups (Figure 1C). Activity of GSSH-R has decreased during storage in all groups. Similar to other tested markers there were no statistically significant differences in the activity of GSSH-R between the groups until day 35. Statistically significant lower activity of GSSH-R was noted in group UHF compare to other groups on the day 35 of storage (Figure 1D).

Discussion

The effect of electromagnetic radiation generated by RFID system on RBC quality has been investigated by several scientific groups. Kozma et al. [13] tested the impact of permanent exposure of 13,56 MHz RFID on potassium, pH, lactate, glucose, free hemoglobin and the hemolysis rate in RBC stored up to 42 days [8]. The observed biochemical changes remained within the normal expected values. They did not detect any obvious adverse effects of RFID on RBC.

The similar investigations conducted Davis et al. [14]. They also observed no notable changes in RBC count, Hb, hematocrit, mean cell volume, RBC morphology score, free Hb, and potassium or percent hemolysis in the 13, 56 MHz RFID tagged RBC units versus control group. These studies had a significant limitation. No in vivo survival studies were performed, therefore, no comment can be made on whether prolonged exposure to radio waves used in RFID system affects RBC recovery or survival.

In our study the similar quality markers for RBC were also not significantly affected by constant exposure of 13, 56 MHz RFID system. However in RBC tagged with UHF RFID labels we observed statistically significant increased hemolysis rate and potassium release on the 35th day of storage compare to units tagged with HF RFID. Wang et al. [15] also showed increase in free hemoglobin, potassium, and sodium in RBC exposed to UHF RFID but the differences to the control group did not reach statistical significance [10]. The clinical significance of the differences found in our study seems doubtful, as significant differences
shown are very minimal. It is less probable that it would really have effect on the clinical outcome of transfusion of components labelled by UHF tags.

These observed changes may be due to the fact that UHF waves can be absorbed by liquids and metals to a greater extent than HF waves. UHF waves are capable of exciting water molecules in blood products to the higher extent than HF waves and may cause increase in the temperature of blood component. It is less probably that the possible increase of temperature induced by radio waves of RFID system directly cause the higher susceptibility of erythrocytes to hemolysis.

Davis et al. [14] indeed proved that use of 13.56 MHz RFID technology is unlikely to have any significant temperature or biologic effects on RBC and plasma units under normal operating conditions [9].

We hypothesize that the energy of radio waves may have an influence on the generation of reactive oxygen species (ROS) and/or activity of antioxidants of erythrocytes. It is known that electromagnetic fields increase free radical activity in cells [11-13]. It is rather due to stimulation of ROS production than inhibiting antioxidant enzyme activity. Radio frequency electromagnetic radiation has been shown to stimulate NADH oxidase in the plasma membranes of mammalian cells which increases ROS formation [14].

Continuous neutralization of ROS by antioxidants may exceed their scavenging capacity and lead to oxidative stress [15]. Studies have demonstrated that radio waves may induce oxidative stress or apoptosis in animal tissues [16-18]. According to our knowledge this is the first study presenting effect of radio waves generated in RFID system on antioxidants and lipid peroxidation in RBC. We found that UHF but not HF waves applied in RFID system caused statistically significant decrease of SOD, GSH-Px, GSH-S-R activity and increase of MDA production in RBC stored for 35 days. It may be at least one of the mechanisms behind increased hemolysis noted in these erythrocytes on 35th day of storage [19-23].

No laboratory findings are available to support the idea that the exposure to high levels of HF radio waves on RBC components could have negative effect on the maintenance of oxidoreductive balance of stored red blood cell components. In contrary UHF RFID tags affect this balance and may lead to eryptosis. The clinical significance of the differences found in our study seems doubtful, as significant differences shown are very minimal. It is less probable that it would really have effect on the clinical outcome of transfusion of components labelled by UHF tags.

In spite of the advantages of UHF over HF waves system the approval of UHF RFID system for blood components needs further studies.

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Conflict of Interest
All authors have disclosed no conflict of interest.

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