How to define and study structural proteins as biopolymer materials

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Abstract
Structural proteins, including silk fibroins, play an important role in shaping the skeletons and structures of cells, tissues, and organisms. The amino acid sequences of structural proteins often show characteristic features, such as a repeating tandem motif, that are notably different from those of functional proteins such as enzymes and antibodies. In recent years, materials composed of or containing structural proteins have been studied and developed as biomedical, apparel, and structural materials. This review outlines the definition of structural proteins, methods for characterizing structural proteins as polymeric materials, and potential applications.

Definition
Proteins are polymers in which the 20 natural amino acids are linked by amide bonds. In addition to the 20 natural amino acids, there are amino acids that are not directly synthesized from ribosomes, such as 3,4-dihydroxyphenylalanine (DOPA), hydroxyproline (Hyp), dityrosine, and selenomethionine, and these compounds are synthesized via posttranslational modifications. These non-ribosomal peptides and amino acids often play an important role in structural and functional proteins. In many cases, structural proteins have a characteristic amino acid sequence that repeats to form a higher-order structure by intermolecular and/or intramolecular hydrogen bonding [1]. One example is the disulfide bond formed between two cysteine residues, which can induce dimerization and hierarchical structures in proteins [2]. However, in general, the amino acid sequence and the structure being formed depend on the type of structural protein, and it is difficult to identify a structural protein based on the sequence or structure alone. In particular, when nonnatural sequences created by gene recombination techniques and chemical synthesis are considered [3], defining a structural protein is even more difficult and complicated. Therefore, in this focus review, I would like to define a structural protein as “a protein that possesses a characteristic amino acid sequence or motif that repeats and forms a skeleton or contributes to the mechanical properties of a living organism, cell, or material” (Fig. 1). Structural proteins can be globular or fibrillar proteins. In most cases, such motifs are expected to form higher-order structures by intermolecular or intramolecular interactions, leading to the expression of physical characteristics. However, considering protein-based amorphous materials, the requirement of a hierarchical structure should not be included in this definition.

Natural structural proteins

To mimic and develop structural protein-like materials, scientists and researchers need to understand the molecular and hierarchical mechanisms influencing natural structural proteins. In this section, several natural structural proteins are reviewed from the perspective of biological polymeric materials.

Spider silk

Many types of spiders are known to produce multiple types of fibers, and some spin up to 7 types (Fig. 2) [4–6]. The thread spun from the major ampullate gland is called dragline and is used as a framework for reticular spider webs and as lifelines. As can be inferred from their role in nature, spider draglines exhibit excellent mechanical
properties and are known as typical “tough spider silk fibers”. In particular, the toughness of these fibers in terms of resistance to breakage due to stretching is significantly superior to that of synthetic polymers and other natural materials because of their relatively high strength and elongation at break [7, 8]; hence, these fibers are expected to be applicable in lightweight structural materials such as automotive parts. The mechanical and physical properties of these fibers are significantly influenced by the deformation rate and humidity [9–11]. Unlike silkworm silk fibers, the physical properties of spider dragline, namely, its supercontraction, are influenced by interactions with water molecules. Recently, proline and diglutamine in the amino acid sequences have been proposed to contribute to supercontraction based on comparative studies of the supercontraction of different spider silks [9]. Spider dragline silk has a tandem sequence of polyalanine-rich and glycine-rich sequences. Polyalanine is a potentially crystalline motif, whereas glycine-rich sequences form amorphous regions. This characteristic repeating motif allows the formation of higher-order structures in which microcrystals comprising beta sheet structures are highly oriented along the fiber axis [12–14].

**Silkworm silk and other silks**

Silk protein is produced not only by spiders but also by animals such as silkworms and scorpions (Fig. 2). The
presence of beta sheet structures as crystal components is similar, but the amino acid sequences and physical properties vary markedly among the different species. Silkworm silk fiber is known to be coated with the adhesive protein sericin, and many studies focusing on the material properties of sericin have been reported [15]. In recent years, a consortium of Asian countries has reevaluated silk as a polymer material and reported the variety of physical properties offered by silkworm silk thread (Fig. 3) [16]. Differences in mechanical and thermal properties can occur due to variations in the hierarchical structure resulting from the amino acid sequence of the silk protein. As silk fibers have been used as sutures, silk is attractive as a biomedical material for regenerative medicine, tissue engineering, and drug delivery [17, 18] because of its excellent biocompatibility, biodegradability, and low cytotoxicity [19, 20]. In addition, some silks have an arginine-glycine-aspartic acid (RGD) sequence, which contributes to cell adhesion, making them useful for biomaterials [21–23].

The relationships among the amino acid sequences, secondary structures, and thermal properties of silkworm silk have been investigated and are summarized in Fig. 4. The predominant secondary structure can be polyglycine II, polyglycine I, or beta sheet structures depending on the composition of alanine and glycine in the copolymer, as these are the amino acids that make up the crystal region of the silk, and the pyrolysis behavior is also influenced by the alanine/glycine composition [24]. This correlation has been confirmed not only in model peptides obtained by copolymerization but also in silkworm-derived natural silk fibers [16]. This result suggests that differences in the partial polypeptide sequences in proteins greatly affect the material’s physical properties, especially for silk materials, for which the thermal and mechanical properties are fully determined by the crystalline sequences. This is recognized as an important finding for the molecular-level design of silk materials with the desired thermophysical properties. In this way, the basic information required for controlling the physical and mechanical properties of structural proteins such as silk is gradually being reported.

Collagen

Collagen is one of the most abundant proteins in mammals and is a major structural component of extracellular matrices. It contains many periodically repeating 3-amino-acid sequences containing Gly. The Gly-containing repeating sequences include ~1400 amino acid residues [25]. The most common tripeptide unit of collagen is Gly-Pro-Hyp, and these three residues form one helical turn. As a structural protein, collagen has excellent biocompatibility and cell adhesion and can promote cell proliferation and differentiation.

Elastin

Consisting mainly of glycine, valine, alanine, and proline residues and possessing a molecular weight of ~66 kDa, elastin is present in connective, vascular, and load-bearing tissues and has very elastic mechanical properties. In rat
skin, elastin plays a mechanical role at small stress with small deformations [26]. The elastic modulus, strength, extensibility, toughness, and resilience are 0.0011 GPa, 0.002 GPa, 150%, 1.6 MJm$^{-3}$ and 90%, respectively [27]. Therefore, elastin is considered an important structural protein and material for scaffolds that require elastic physical properties in cell culture. However, elastin is poorly soluble in aqueous solutions, making it difficult to process, and this is a common problem among structural proteins. In addition, since elastin is difficult to obtain, synthesize, and produce in large quantities, reports on biomaterials composed of elastin alone are very limited compared with reports on scaffolds composed of other structural proteins. A scaffold of elastin derived from bovine ligament has been reported, and its mechanical properties have been studied, with a focus on its elastic modulus [28]. The elasticity of elastin makes it an effective additive for improving collagen-based tissue engineering vessels, showing that elastin hybrid composites offer unique mechanical properties such as significant elasticity, indicating that elastin is a useful structural protein for preparing elastic structural materials [29].

**Resilin**

Resilin, a structural protein, is an entropy elastic material (rubber) found in biological structures that require energy storage and long-range elasticity, and it exhibits 300–400% elongation at break, low solubility, and thermal stability up to ~140 °C [30]. Resilience (the ratio of the energy elastically returned to the added energy) of resilin is ~92% due to covalent crosslinking between tyrosine residues [31]. The mechanical properties of elastin are similar to those of resilin, but resilin shows higher resilience and lower modulus. Resilin can be found in various biological joints, such as the vein joints of dragonfly wings [32]; however, natural resilin is not widely available for biological and physical characterization (Fig. 5). Therefore, by partially cloning and expressing the Drosophila gene that has been identified to encode resilin, a recombinant protein for resilin was successfully designed and obtained, and a rubber-like material could be prepared by rapid photochemical crosslinking [33]. This crosslinking was achieved by peroxidase, which

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**Fig. 4** Comparison of amino acid sequences, thermophysical properties, and secondary structures of the crystalline sequences of silkworm silk protein. a–c Modified with permission from refs. [16, 24] Copyright 2017 American Chemical Society

**Fig. 5** Distribution of resilin in the vein joints of the Rhinocypa fenestrella wing. a Resilin present on the dorsal side of the wing. The purple dots indicate resilin. b Fluorescence microscopy (left panels) and scanning electron microscopy images (right panels) of the selected joints. Adopted from ref. [32]
catalyzes dityrosine formation. The recombinant resilin proteins were expressed and evaluated as biomaterials [34–36]. Due to the attractive physical characteristics of resilin, combinations of resilin with other proteins have also been investigated [37, 38]. Elastomers composed of biomolecules, including resilin, are often photochemically or chemically crosslinked. These biomaterials behave as rubber-like materials and exhibit high elasticity with small strains, and they are expected to be applicable as new muscle-mimetic biomaterials. On the other hand, examples of research on resilin are limited compared to the number of studies on elastin, as many researchers have studied the use of the latter as a biologically derived elastomer.

Keratin

Keratin is the main structural protein that forms the hair, wool, feathers, nails, and horns of many types of animals. This protein has a high content of cysteine (7–20% of the total amino acid residues), which is known to form intramolecular and intermolecular disulfide bonds [39, 40]. It has been reported that α-keratin, which has a helical structure, can be reduced to form β-keratin upon stretching treatment, and this structural change affects the mechanical, thermal, and chemical properties of the material [41]. The disulfide bonds in a tissue containing keratin, such as wool, can be disrupted by reductive treatment, affording highly soluble keratin. Keratin can be extracted with an aqueous solution containing urea, mercaptanol, and sodium dodecyl sulfate as a surfactant, and then, insoluble films and biodegradable sponge scaffolds can be prepared [42–44].

Reflectin

Reflectin has been found in certain cephalopods (especially squid), including Euprymna scolopes and Doryteuthis opalescens [45]. Squid can change their body color to camouflage themselves, that is, mimic their surroundings, through the function of reflectin, but research on reflectin protein is very limited, as is our understanding of the molecular mechanism. Reflectin is speculated to have random coil and beta barrels as the main motifs [46], but the structure has not been fully elucidated. Protein families that exist in cephalopod pigment cells (rainbow cells, a type of pigment cell that changes the wavelength of reflected light) and white pigment vesicles and contribute to camouflage function have been recently reported. The amino acid sequence of reflectin is rich in aromatic and sulfur-containing amino acids, and these motifs are used to refract incident light in certain environments in certain cephalopods [47, 48]. However, not all the molecular mechanisms have been fully clarified. Aliivibrio fischeri, a marine luminescent bacterium, possesses the reflectin gene of cephalopods due to horizontal gene transmission [49]. Recently, several material scientists have been interested in reflectin as a new functional biopolymer and have reported novel optical biomaterials containing recombinant reflectin [50].

Synthesis and production

The synthesis methods suitable for preparing structural proteins are dependent on many factors, such as their amino acid sequence, solubility in a given solvent, and toxicity to the host cells used for production. In this section, the current synthesis methods are outlined by taking spider silk protein as an example, as it has been studied in a relatively wide range of host organisms. Spiders are carnivorous and cannibalistic, and hence, it is extremely difficult to raise and culture them. Thus, it is enormously challenging to obtain natural spider silk from spiders in bulk. Therefore, in attempts to artificially synthesize spider silk, various organisms, such as plants, goats, Escherichia coli, yeast, and photosynthetic bacteria as well as animals that construct cocoons, have been used as hosts to synthesize silk protein, the main component of spider fibers. Here, the production and synthesis methods are divided into biological synthesis and chemical synthesis methods.

Biological synthesis

Escherichia coli

E. coli is the most convenient and widely used protein synthesis/production host. In research institutions, such as universities, many researchers use this organism in the laboratory. Accordingly, purification methods have been established at the laboratory level (small scale). The synthesis of recombinant proteins such as recombinant silk proteins can be broadly defined in two major steps: (1) design, construction, and cloning of the genes and (2) expression and purification of the protein polymers. E. coli strains such as BL21, JM109, SG13009 pREP4, and RY-3041 have been used as host systems, and protein yields were inversely correlated with the size of the synthetic genes. Synthetic genes encoding dragline silk from Trichonephila clavipes (formally Nephila clavipes) have been successfully constructed, cloned, and expressed [51]. The synthesis of these recombinant genes is based on the repetitive sequences found in native dragline silk genes. The methods for construction of these repeats were previously reported, using smaller oligonucleotide repeats and subsequent multimerization by traditional methods with restriction enzymes and ligase. Thus, the size and sequence of the protein generated can be controlled by the primary
sequence synthesized as oligonucleotides. The use of synthetic gene technology to control silk protein size allows for the study of relationships between sequence length and structure-function, along with the study of novel compositions. In general, expression levels obtained from synthetic protein genes are low, with yields of purified protein from 1 to 10 mg/L, usually representing less than 5% of the total protein in the cell, depending upon the size of the protein of interest [52–55]. By optimizing the sequence and culture conditions, the productivity of recombinant proteins can be drastically improved.

**Yeast**

A wide variety of yeasts including *Saccharomyces cerevisiae* have been widely used in the fermentation industry, and many companies prefer to use these microbes in fermentation production. Spider silk production in yeast has a long history; for example, recombinant spider silk protein was successfully expressed in the methylotrophic yeast *Pichia pastoris* [56, 57]. *P. pastoris* was reported to exhibit relatively high yield and productivity, with 300–1000 mg/L recombinant spider silk proteins.

**Transgenic plants and animals**

Tobacco and potato have been used as host plants to produce spider silk proteins [58, 59]. However, the synthesis of the full-length spider silk protein, namely, native-sized spidroins, from plants remains challenging [60]. In addition to plants, transgenic goats have been designed to synthesize recombinant spider silk proteins in their milk. Furthermore, the silkworm *B. mori* was genetically modified to synthesize recombinant spider silk proteins [61].

**Photosynthetic bacteria**

As a “sustainable cell factory”, photosynthetic bacteria are attractive biological hosts to capture and fix CO₂ and N₂ (Fig. 6) [62]. The author has been interested in purple photosynthetic bacteria because of their biopolymer productivity and their ability to transform technological platforms [63–68]. Silk proteins were successfully synthesized by using recombinant nonsulfur photosynthetic bacteria in the marine environment. Importantly, through photosynthesis and nitrogen fixation processes, structural proteins can be synthesized under marine conditions by utilizing

![Fig. 6 Putative metabolic pathways by which marine purple photosynthetic bacteria convert CO₂ and N₂ to amino acids, showing the great potential of marine purple photosynthetic bacteria as a platform for synthesizing structural proteins](image-url)
low-cost and abundant renewable resources such as light (energy), CO₂ (carbon source), and N₂ (nitrogen source).

**Chemical synthesis**

**Solid-phase and liquid-phase methods**

Solid-phase synthesis is one of the most common techniques used to synthesize relatively short polypeptides (peptides of up to ~50 amino acids) [69, 70]. From a solid surface modified with an amino group, the amino acid chain is extended one residue at a time by dehydration condensation to obtain the target peptide/polypeptide, and then, the solid surface is cleaved to obtain the target peptide/polypeptide. In the condensation reaction, the amino acid side chain must be protected so that it does not react. In general, the method is not scalable, and it is considered unsuitable for synthesizing proteins and polypeptides in bulk. Reactions in a solvent that do not use a solid surface are called liquid-phase reactions, and the basic mechanism is similar to that of solid-phase synthesis [71]. The liquid-phase method is relatively suitable for large-scale synthesis.

**Native chemical ligation**

In the solid-phase and liquid-phase methods described above, the synthesis efficiency decreases as the molecular weight of the target increases. To compensate for this technical challenge, native chemical ligation is considered a strong complementary reaction [72, 73]. With native chemical ligation, a linked polypeptide can be obtained by mixing a polypeptide with a thioester at the C terminus and a polypeptide with a cysteine at the N terminus, and these precursors can be synthesized by solid-phase or liquid-phase methods. Thus, polypeptides and proteins with more than 200 residues can be chemically synthesized [74]. In recent years, systems other than those based on cysteine have been reported, and their versatility is increasing [75].

**α-Amino acid-N-carboxylic anhydride (NCA) ring-opening polymerization**

NCA has high electrophilicity and can efficiently afford polypeptides by living anion ring-opening polymerizations using an amine as an initiator [76]. Various polypeptides can be synthesized using an appropriate initiator amine [77]. Currently, the NCA method is not used industrially because a highly toxic reagent such as phosgene or triphosgene is necessary for the preparation of NCA. In recent years, phosgene-free syntheses of NCA have been reported and used to synthesize polypeptides [78, 79]. However, sequence control is still technically difficult, and protein sequences are not easy to mimic.

**Chemoenzymatic polymerization**

Chemoenzymatic polymerization allows the synthesis of peptides in a relatively short time and with high yield and atom economy [80, 81]. In chemoenzymatic polymerizations for peptide synthesis, amino acid derivatives such as amino acid esters are polymerized via protease-catalyzed aminolysis (Fig. 7) [82, 83]. In addition to natural amino acids [84–86], unnatural amino acids, branched structures, and blocky sequences can be easily introduced [87–90]. By copolymerizing 2-aminoisobutyric acid and nylon units with peptides, the secondary structure and thermal properties of the resulting material can be dramatically altered [90–92]. The disadvantage of this method is that it is difficult to control the molecular weight and sequence of the polymer. At present, the application of chemoenzymatic polymerization to the synthesis of polypeptides is still in development.

**Synthetic protein material products in the industry**

Protein-based materials have been well reviewed by several groups [93–95]. Generally, the mechanical/physical properties of structural proteins in nature are expected in their regenerated materials (Table 1). However, it is not easy to reproduce the native physical properties and biological functions of a structural protein using regenerated or recombinant proteins, even though structural proteins can be synthesized by various methods. The reason is that the hierarchical structure and natural biological arrangement are difficult to reproduce by artificial processes. The material processing method needs to be considered and optimized to achieve the desired physical/biological properties of the structural protein material. Here, several examples of structural proteins that have already been commercialized or produced in bulk as a material as well as materials containing a structural protein are listed.

**Bolt Threads Microsilk™**

Although not disclosed in detail, Bolt Threads (CA, USA) has developed artificial silk fiber, Microsilk™, by a recombinant protein based on the major ampullate spidroin (Fig. 8). The Bolt Threads Microsilk™ Tie is composed of 100% Microsilk™. In 2017, 50 ties were sold at $314 each, and they sold out immediately. Best Made Co. x Bolt Threads Cap of Courage was released in 2017 with 100 colors and 10 patterns at $198 each, and they sold out immediately as well. These hats are a composite of Microsilk™ and American Rambouillet wool. The Stella McCartney x Bolt Threads Gold Microsilk™ Dress is made of 100% Microsilk™ and was designed by Stella McCartney. It was presented and
exhibited at “Items: Is Fashion Modern?” at the Museum of Contemporary Art in New York.

**Spiber Inc. QMONOS™ and Brewed Protein™**

Based on spider silk-derived proteins, Spiber Inc. developed artificial silk fibers with various molecular designs, which can be used for the production of apparel and materials (Fig. 9). Moon Parka™ and T-shirts made of structural protein fibers and textiles were commercialized in 2019. The company also demonstrated silk resin, film, and other silk materials, although these materials have not yet been commercialized as of April 2020 (Fig. 9).

**Casein plastic**

Casein is a phosphorylated protein present in milk and cheese and is not generally regarded as a structural...
The colloidal casein in milk solids is composed of not a single type of protein but a combination of α-casein, β-casein, and κ-casein. Focusing on the amino acid composition of casein, most of the serine residues are phosphorylated, and this is regarded as a characteristic structural feature of casein. However, in recent years, biomaterials using casein as structural and/or bulk materials have been developed and produced commercially. Casein plastic was invented in Germany in 1898. Acid is added to milk to solidify casein into a material. The precipitated casein becomes a thermoplastic with an appearance similar to that of ivory. This material is called casein plastic or lactic casein and is used industrially for impressions, clothing buttons, etc. It is not actually a thermoplastic, but it is recognized as a biodegradable, biologically derived material.

Promix

Promix, a fiber made by graft polymerization of acrylonitrile on casein (Fig. 10), was developed by Toyobo Co., Ltd, in the 1970s. The texture is similar to that of silk. Materials that combine Promix and bioplastics such as polylactic acid are also commercially available.

Plant protein fibers

Fibers can be prepared from proteins extracted from plants that have a high protein content, such as soybeans (Fig. 10). Textiles have been developed from peanuts, corn and soybeans. These proteins are difficult to fibrillate as a single substance and are commercialized by compounding with polyester or nylon. For example, a composite fiber was made of 20% soybean protein, 20% nylon, and 60% polyester.

Detection of structural proteins in polymeric materials

There are various natural and synthetic structural proteins and structural protein-like polymers, as introduced above. The content of structural proteins is always important when considering the biomass origin or biodegradability of a material. Furthermore, structural protein quantification is necessary for determining the content of structural proteins in blends and/or composite materials for commercialization. The methods for quantifying structural proteins are the same as those typically used for protein quantification; however, a
suitable method must be identified based on the structure, shape, morphology, and size of the material to be measured.

**Quantification and qualification**

When the whole material is composed of proteins, there is no need to quantify the protein content. However, when in composite materials with other materials, such as polymer resin, the protein content in the material must be determined. If the structural protein is separable from other resins in a composite/blend material, the structural protein can be quantified by the weight ratio (Fig. 11); i.e., the separated structural protein parts are quantified by weight. If the structural protein parts are inseparable, the structural protein needs to be extracted for the content to be quantified.

To extract the protein from a composite/blend material, a pretreatment is necessary prior to extraction. The material is first finely pulverized and subsequently treated with a reducing agent in a denaturing buffer. Typical examples of reducing agents are 10 mM dithiothreitol (DTT), mercaptoethanol, and tris(2-carboxyethyl)phosphine, whereas examples of denaturing buffer are 6 M guanidine hydrochloride, 8 M urea, and formic acid. However, each reducing agent and denaturing buffer combination must be considered with caution because dangerous reactions can occur such as that between urea and chlorine bleach.

To quantify a protein, namely, to determine the concentration of protein in a solution, the UV method, BCA method, Bradford method, etc., can be used. The UV method involves detecting the absorption at ~280 nm, which is the major absorption of tyrosine and tryptophan, but this strategy is not suitable for the quantification of structural proteins because it depends on the tyrosine and tryptophan content of the protein. The Biuret method is based on the detection of Cu+, which is produced when Cu2+ reacts with a protein under alkaline conditions, is an appropriate technique, as are the Lowry method and BCA method, which are advanced variations of the Biuret method. However, the Lowry method is easily affected by a reducing agent such as DTT and is not suitable for this purpose considering the extraction conditions for structural proteins. On the other hand, the BCA method is inhibited in the presence of ~1 mM DTT, ~4 M guanidine hydrochloride, and ~3 M urea but can be used for quantification if the structural protein solution is sufficiently dilute. The Bradford method is conducted under acidic conditions, while the BCA method uses alkaline conditions. Many structural proteins have lower solubility under acidic conditions, making BCA quantification under alkaline conditions the more suitable method. When using the BCA method, both the denaturation buffer and the protein extract must be sufficiently diluted prior to use.

**Confirmation of the presence of protein**

To confirm that the test material contains some amount of protein, the amino acids, and amide bonds of proteins must be detected by the ninhydrin method [97–99]. In the ninhydrin method, the sample is dissolved in a protein denaturing solution, and the resulting solution is used for amino acid analysis. For amino acid composition analysis, a protein sample is hydrolyzed with hydrochloric acid, and each amino acid is quantified by HPLC. Using this strategy, 17 types of amino acids can be confirmed because tryptophan, asparagine, and glutamine cannot be quantified individually. The contents of asparagine and glutamine are combined with those of aspartic acid and glutamic acid, respectively, due to the hydrolysis of asparagine and glutamine during the sample preparation process. Tryptophan is also degraded during preparation, and hence, other protocols for determining tryptophan are needed. When posttranslationally modified peptides and amino acids, such as Hyp, DOPA, and dityrosine, in structural proteins need to be quantified, standard substances corresponding to the target amino acids must be added to the mixture [100].

**Confirmation of structural proteins among various proteins**

Pure structural protein can be quantified by the general protein quantification method, as described above. However, if structural proteins and other proteins are mixed, they cannot be distinguished by the abovementioned method. To not only quantify a protein but also confirm that it is a structural protein, after treating the extracted protein
with a degrading enzyme such as trypsin or separating the protein by sodium dodecyl sulfate–polyacrylamide gel electrophoresis and extracting the protein from the gel, matrix-assisted laser desorption/ionization time-of-flight mass spectrometry should be performed, and databases can be searched for possible sequences. This method requires the protein to have a known sequence, and it is difficult to quantify a novel structural protein prepared by chemical synthesis or genetic recombination technology using this strategy. In addition, shotgun proteomics as well as targeted proteomics using selected reaction monitoring and data-independent acquisition methods are alternative methods. However, modification with synthetic polymers will significantly prevent those characterizations. Thus, in protein quantification, it is generally difficult to specifically quantify proteins with unknown structures. Furthermore, impurity (contaminant) proteins contained in the protein materials synthesized by microbial fermentation must be considered in these analyses.

Summary and future perspective

In this focus review, we focused on structural proteins and broadly outlined them in terms of their definitions, properties, and syntheses as well as their potential as materials and quantification techniques. As mentioned at the beginning of this review, although structural proteins such as silk and elastin are used as structural materials in nature, they have not been put into practical use as bulk-scale structural materials for human use. The challenges of using proteins and polypeptides as structural materials are multifaceted and include their material design, synthesis, purification, and processing. In the future, the author hopes that more scientists will enter this field and establish innovative techniques for molecular design, synthesis, material design, and processing to achieve structural proteins with optimal physical and biological properties.

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Compliance with ethical standards

Conflict of interest The author declares that he has no conflict of interest.

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Keiji Numata earned his Ph.D. (2007) with a thesis centered on enzymatic degradation and synthesis with hydrolases of biopolymers, especially poly(hydroxyalkanoate), under the supervision of Prof. Yoshiharu Doi, Tokyo Institute of Technology. His Ph.D. thesis includes works on enzymatic polymerization to synthesize branched biopolymers, which has been performed in Royal Institute of Technology (Sweden) under the supervisions of Prof. Ann-Christine Albertsson and Prof. Anna Finne-Wistrand. He worked as a JSPS Postdoctoral Fellow for Research Abroad at Tufts University (USA) where he studied biosynthesis of silk-based polymers via bacterial pathways as well as silk-based gene carriers in the laboratory of Prof. David L. Kaplan. He moved to RIKEN as a Senior Scientist in 2010 to start up a laboratory to investigate biosyntheses and material design of structural proteins and poly(amino acid). He has been a Team Leader (PI) of the lab since 2012 and Research Director, JST-ERATO Numata Organelle Reaction Cluster Project since 2016. In 2020, he moved to Department of Material Chemistry, Kyoto University, as a full professor. He received 2020 ACS Macro Letters/Biomacromolecules/Macromolecules Young Investigator Award, American Chemical Society (2020), SPSJ Asahi Kasei Award (2019), Award for Encouragement of Research, Japanese Society for Plant Cell and Molecular Biology, Japan (2019), Bio-Environmental Polymer Society Outstanding Young Scientist Award, USA (2018), The Young Scientists’ Prize for Minister of MEXT, Japan (2018), and so on. He was appointed as an associate editor of Polymer Journal (2018–2020) and is currently an associate editor of ACS Biomaterials Science and Engineering.