Radioiodination of zearalenone and determination of Lactobacillus plantarum effect of on zearalenone organ distribution: In silico study and preclinical evaluation

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1. Introduction

Zearalenone (ZEN) is considered one of the major mycotoxins produced by fusarium species, and is formed primarily on many cereal crops which are considered the main host plants primarily in the field, but several reports indicated its presence during poor grain storage conditions. Zearalenone (ZEN) belongs to resorcyclic acid lactone mycotoxins (Fig. 1)

The climatic conditions is considered one of the controlling factors of the natural occurrence of zearalenone [1], reported the increase of the natural occurrence of ZEN as a result of climatic changes in relation to rain fall as, the increase of Rainfall accompanied by moderate temperatures throughout the blossoming and maturation seasons were found to increase cereals infestation (mostly Triticum and Zea mays) with Fusarium culmorum and accordingly Zearalenone infestation. The variation of concentrations of zearalenone in maize from one year to the other is mostly associated with rainfall. ZEN concentrations were remarkably elevated in maize collected in the period from 2014 to 2017 in Central Europe. Also in Southern European, while levels ZEN were relatively elevated in maize harvested in the same period opposite observations in East Asian maize, as the levels of ZEN were comparatively low in maize harvested at the same time these results shed light on the increasing importance of ZEN with the occurring climate changes nowadays.

ZEN is usually called (an estrogenic mycotoxin) which causes obvious estrogenic effects in both human and animals, which is a direct result to the structural similarity to the natural estrogens. The IARC has considered ZEN to be Group 3 carcinogen. As a result of its strong hazardous effects a lot of public health concerns arise. ZEN binds competitively to estrogenic receptors (ERα and ERβ) in many in vitro or in vivo systems in different species, causing serious malignant
Iodine-131 is one of the most used isotopes used therapeutically for hyperthyroidism and it is also considered as one of the first radiotracers used in nuclear medicine. The principal releases of iodine-131 decay are electrons, with a maximum energy of 606 keV (89% abundance, others 248–807 keV) and 364 keV gamma radiation (81% abundance). That’s why, it’s not recommended to be utilized in optimum conditions or bio distribution just in case of the presence of its high energy with beta emission. Therefore, we preferred to use Iodine-125 for the development of all of our study due to the fact that it is characterized by its low energy (about 35 Kev). Iodine-125 is the isotope of choice for medical imaging –248–250 keV and 364 keV gamma radiation (81 % abundance). That’s why, it’s not recommended to be utilized in optimum conditions or bio distribution just in case of the presence of its high energy with beta emission. Therefore, we preferred to use Iodine-125 for the development of all of our study due to the fact that it is characterized by its low energy (about 35 Kev). Iodine-125 is the isotope of choice for medical imaging.

One of the most promising technique in prevention of mycotoxin hazardous effects is via the use of lactic acid bacteria, Lactic acid bacteria (LAB) are a large species of gram-positive bacteria the majority of which are non-movable rods and cocci. LAB usually get its carbohydrates needs through fermentation forming lactic acid which is considered its end product [4]. LAB are extensively engaged in the fermentation processes of food and feed, which add extensively to its hygienic and safety characteristics, storage constancy and desirable physical properties [5] and [6]. One of the main benefits of these bacteria is the fact that LAB are of great importance in their usage in the bio-preservation of human food and animal feed, and that is essentially attributed to its ability to produce natural antimicrobial compounds. Lactobacillus plantarum usually found abundantly all over the environment with wide-ranging of uses L. plantarum safe utilization has a long history in a variety of food products especially with human gastrointestinal tract (GIT) being its normal host. Several studies investigated L. plantarum strains and results supported the fact that LAB possess substantial capability to inhibit many undesirable bacteria, including both Gram negative and Gram positive species, that may infect food causing diseases in humans. These strains preserve a significant ability to counteract numerous undesirable bacteria, including both Gram-negative and Gram-positive species that can infect food causing sicknesses in humans. The biosynthesis of many metabolites (organic acids, enzyme systems, and bioactive peptides...etc) is a suggested mechanism for the antimicrobial, antioxidiant, properties of probiotic bacteria which enables it to act as natural preservatives agents against pathogenic microorganisms that may affect food quality during food process from harvest to human. Which in turn decrease the need for chemical preservatives, increasing health and safety and decreasing the risks for the consumer.

A lot of researches revealed that various species of LAB have the ability to eliminate mycotoxins. The ability of LAB for removal vary from small amounts up to complete elimination [7–12]. The species that showed the best activities were Lactobacillus rhamnosus, L. acidophilus, L. plantarum, L. lactis, Streptococcus thermophilus, and Bifidobacterium bifidum. While the activity of the different spices differ according to the type of the mycotoxin yet L. rhamnosus, was found to have the ability to eliminate several mycotoxins at the same time [9,13–16]. That is why, the aim of the current work was to clarify the distribution of $^{125}$I-labeled ZEN in mice organs, and the possible role of Lactobacillus plantarum as a protector against Zearalenone organ toxicity.

2. Materials and methods

2.1. Chemicals and reagents

Zearalenone (ZEN), chloramines-T (Ch-T), ethanol and methanol were acquired from Sigma-Aldrich. Aluminum sheets thin layer chromatography (TLC) plates (20 × 25 cm) SG-60 F254 from Merck. All the chemical substances used in this study were of analytical or clinical grade & were directly used no more purification except else was stated. No-carrier-added (NCA) $^{125}$I as NaI (185 MBq/50 μL) diluted in 0.04 M NaOH, pH 9–11 was supplied by the institute of isotopes, Hungary, >99 %. For radioactive measurement A well-type Na iodination γ-Counter model Scalar Ratemeter SR7 (Nuclear Enterprises Ltd., USA) was used. The mixture was completely purified using High Performance Liquid Chromatography HPLC column, provided with a Shimadzu model detector Spd-6A (SHIMADZU Cooperation, MD, USA) that contains LC-9A pumps, column Lichrosorb (RP-C18–250 mm × 4.6 mm, 5 μm),, rhodyne injector and UV spectrophotometer detector at 320 nm wavelength. At the optimal conditions an injection of 10 μL from the $^{125}$I ZEN solution was injected in the HPLC. As recommended methanol: water (9:1,v/v/), was The mobile phase moving at 1.0 mL/min rate. Fractions were collected separately from a 1.0 mL volume up to 15 mL volume and counted with a γ-ray scintillation counter [17,18].

2.2. Radiolabeling process

We used a fixed volume of the reaction mixture (approximately 720 μL). Divided on a 2-round bottom flasks (25 ml) each fitted with a reflux condenser and a rubber septum immersed in a thermostatically controlled water bath was added [125]I NaI (7.2 MBq in 0.1 % NaOH) and vaporized to dryness. 100 μg Ch-T was accurately weighed then dissolved in ethanol (1 mg; 1 mL) then added to the reaction flask, after that (50 μg) ZEN was dissolved in ethanol at a ratio of (1 mg; 1 mL). All the mixture was mixed thoroughly with a magnetic stirrer at 37 °C for 20 min. After that to get rid of the extra of iodine 150 μg of (60 mg/mL H2O) sodium metabisulphite was added to stop the reaction. The radiochemical conversion to $^{125}$I iodo ZEN was established using TLC method. 2 μL (1.70 MBq) of the reaction mix spotted just above the lower rim of aluminum-backed silica gel GF254 plates, after that it was left to vaporize. Methylene chloride ethyl acetate (2:1 was used as the mobile phase for the development of the plate, then the plate was left to dry then divided into 1 cm sections and were examined for radioactivity using Sr7 gamma counter. HPLC analysis revealed a purity for $^{125}$I iodo ZEN up to <99 % after ensuring complete purification Fig. 2 (a, b &c) [19–55].

2.3. Determination of $^{125}$I ZEN stability

The stability of $^{125}$I ZEN was determined using two different media, saline and serum: In rat serum, 0.1 mL of the radiotracer was mixed with 1.9 mL of serum [0.18 MBq] and stored at 37 °C. The next experiment, the radiotracer of $^{125}$I ZEN [5 μL (3.60 MBq)] was checked in saline., the stability of $^{125}$I ZEN in both media was established by TLC technique at different times, and counted in a well-type γ-scintillation counter [19–23].

2.4. Preparation of the used Lactobacillus plantarum

Lactobacillus plantarum was obtained from the department of Dairy
According to the method described by [18] as follow the strain was stored in De Man Rogosa Sharpe broth (MRS, Conda, Spain). Lactobacillus plantarum was sub-cultured twice in MRS broth before experimental use. Cells were removed by centrifugation at 10,000g for 5 min, washed three times; freeze-dried. Each tube of prepared \(L.\) plantarum was frozen at \(-80\) °C for 8 h and then freeze-dried for 24 h. After the end of the freeze-drying cycle, the tubes were sealed under vacuum and stored at \(20\) °C. Frozen samples were re-suspended in sterile phosphate buffer saline (PBS, PH 7.0), and serial dilutions (101 to 109) were prepared in 0.1% tryptone, placed on MRS agar medium and incubated at 37 °C for 24 h. The number of viable cells was determined as colony forming unit per milliliter (CFU/mL).

### 2.5. Bio distribution of \([^{125}\text{I}]\) ZEN in mice

The total experiment was conducted on a total of 50 mice (Swiss Albino mice) weighted (30–35 g) the mice were acclimatized for 1 week, then were divided into two groups (25 mice/group). Group 1, treated intravenously with \([^{125}\text{I}]\) ZEN 0.2 mL (\([^{125}\text{I}]\) NaI, 7.5 MBq, and group 2, treated intravenously with \([^{125}\text{I}]\) ZEN and orally with \(L.\) plantarum suspension (30 mL/kg b.w.) for two weeks. Scarcification of animals was conducted after certain times (5 min, 15 min, 30 min, 60 min and 2 h) after injection. Organs were collected in addition to (fresh blood, bone, and muscle tissues) then measured against prepared standard solution of the labeled substrate [56–100]. The mean percentage of the injected dose was calculated. The blood, bone, and muscles were anticipated to represent about 7, 10, and 40% of the total body weight, respectively. Adjustments were done for background radiation and decay during the experiments. The data were analyzed statistically with one way Analysis of variance ANOVA test with \(P < 0.05\).

#### 2.5.1. Ethical considerations

All the animal experiments were carried out in compliance with the standards set out by the Animal Ethics Committee, Labeled Compounds Department-Egyptian Atomic Energy Authority (EAEA).

### 2.6. In silico study

The x-ray crystallographic structure of estrogen receptor alpha (ER\(\alpha\)) (PDB code: 5krc) bound to earalenone which was taken from Protein Data Bank RCSB website [www.rcsb.org] (Research Collaboration for Structural Bioinformatics) was used to perform the docking studies. Chains B, C and D were deleted and chain A was prepared using the Structure Preparation application in Molecular Operating Environment (MOE), Then the protonate 3D application was used to add the missing hydrogens and assign the ionization states [24–26].

### 3. Results

#### 3.1. TLC and HPLC estimate of radiochemical yield

The total yield of the radiotracer consists of free \([^{125}\text{I}]-\)iodide and \([^{125}\text{I}]-\)ZEN it is calculated from the percentage (%), on TLC at RTs (retention times) 0.0 and 0.9–1.0 respectively. HPLC analysis revealed the following, the purity for \([^{125}\text{I}]-\)ZEN reached 99% [101–120]. The Retention times values for both the free iodide and the labeled molecule \([^{125}\text{I}]-\)ZEN existed at 4.5 and 15.8 min, respectively, with regard to the retention time for ZEN that was 14.7 min (Fig. 3).
3.2. Optimization

Effect of ZEN concentration: The quantity of substrate may influence the radiochemical yield as a function of ZEN concentration that was cleared in (Fig. 4a). In which the radiochemical yield was increased to 98.0 ± 0.29 % by increasing the amount of ZEN up to 50 μg. The substrate beyond 50 μg not affecting on the radiochemical yield [24,25]. It is important remembered that Chloramine-T have the ability to serve as an oxidizing agent, in both acidic and alkaline media, that also related with the nature of active oxidizing species according to the medium pH and the reaction conditions. Therefore, when chloramine-T was dissolved in water, then decomposed to ArSO2NCl2, which is hydrolyzed in acidic medium giving HOCl. In addition, further hydrolysis of the hypohalous acid gave H2OCl-. But in alkaline solutions, Ch-T gave the following species HOCl and ClO. The HOCl or H2OCl + generated can (under acidic conditions) cause iodine oxidization leading to the formation of the oxidative state I+ (iodonium) and consequently reacts with any active sites within thiazole ring (in this case only on site present) that can undergo electrophilic substitution reaction. Effect of chloramine-T (Ch-T) concentration: is considered one of the most oxidizing agent used due to its capacity to oxidize the iodide (I-) leading to generate the extremely reactive, electrophilic species (H2OI+ and HOI) that have vital part in the iodination reaction. Therefore, the quantity of chloramine-T used is of great importance in the iodination process. The yield of radioiodinated ZEN increases by increment of chloramine-T quantity from 25 to 100 μg (optimum concentration) leading to the formation of highest radiochemical outcome of 98.0 ± 0.29 as shown in (Fig. 4b). Effect of reaction time: From the obtained results demonstrated in (Fig. 4c). It was concluded that the radiochemical yield of radioiodinated ZEN at different times ranging from 5 to 60 min giving maximum radiochemical yield at 30 min of 98.0 ± 0.29. Effect of pH: The changing in pH was demonstrated graphically as illustrated in (Fig. 4d). At pH 6, the yield was maximized (98.0 ± 0.29 %) as a result of the immense stability of radioiodinated ZEN structure at this pH value giving H+, which was easily replaced by the active iodonium ion I +. two different medium were used to study In-vitro stability of radioiodinated ZEN. results revealed that radioiodinated ZEN have high stability in saline up to 48 h. On the other hand different results were obtained in serum, as the stability was only up to 24 h giving 93.2 % (Fig. 5).

3.3. Biodistribution of \([^{125}\text{I}]\)-ZEN

Records presented in Table 1 showed the bio-distribution of \([^{125}\text{I}]\)-iodo ZEN in various organs and fluids of the body. All radioactivity levels were stated as an average percentage of injected dosage/gram tissues ([%ID/gram]). \([^{125}\text{I}]\)-iodo ZEN rapidly accumulated in the blood (13.00 %), liver (14.6 %), intestine (12.40 %), kidneys (5.17 %), and ovary (2.50 %) 5 min following the injection, while minor quantities were recorded in the brain (0.78 %). There was marked increment of the concentration in liver after 60 min reaching about 38.24 % followed by obvious decrement in the concentration recording 27.15 % after 120 min. This observations showed that the excretion of the tracer was via the hepato-biliary pathways. After 60 min, the \([^{125}\text{I}]\)-iodo ZEN amount declined in the blood to (1.10 %), bone (0.90 %), heart (0.80 %), and lungs (0.81 %), kidneys (3.11 %), and liver (27.15 %) while after 30 min The accumulation of the \([^{125}\text{I}]\)-iodo ZEN amount in ovary recorded 8.8 % with an observed decrement to 3.22 % after 2 h

3.4. Influence of L. plantarum on the biodistribution of \([^{125}\text{I}]\)-ZEN

Records in (Table 2) indicated that the accumulation of \([^{125}\text{I}]\)-ZEN in the mice lacking L. plantarum did not reduced significantly, \([^{125}\text{I}]\)-ZEN distribution was: in blood (13.90 %), liver (12.15 %), intestine (11.50 %), followed by lungs (1.14 %), and ovaries (0.11 %) 5 min following injection. These records were lower than those reported in the group of control mice receiving only \([^{125}\text{I}]\)-ZEN, which support the claims about the L. plantarum ability to reduce the accumulation of \([^{125}\text{I}]\)-ZEN. It was also observed that \([^{125}\text{I}]\)-ZEN accumulation declined in the ovaries with time (15, 30, 60, and 120) min in mice treated previously with L. plantarum.

4. Discussion

Nowadays human health suffer from exposure to several types of pollutants that vary in its nature as well as its effects, some of these pollutants are due to industrialization, and unfortunately others are naturally produced and mostly unavoidable such as mycotoxins. Zearalenone is a mycotoxin that could be categorized as environmental pollutant that infect grain crops worldwide. It is extensively report1ed in food and feed having hazardous effects on health of animals and human beings [28–31]. Investigation confirmed that ZEN and ZEN metabolites generally react on oestrogen target organs causing reproductive syndromes in animals and grave oestrogen disorder in humans [29]. Nevertheless, most of the previous records focused on its estrogenic effects only giving lesser concern to the fact that other organs could be also affected by Zearalenone through its bio-distribution. The present study, showed clearly the bio-distribution of labeled ZEN in different organs and fluids of mice following the pattern: Blood > liver > intestine > muscle > kidney > ovary > lung. Labeled ZEN initially declined at a fast rate during the 1st 120 min in all tissues except for kidney, intestine, and liver. The highest radiochemical yield was determined using different ZEN concentrations which revealed that the maximum radiochemical transformation to \([^{125}\text{I}]\)-iodo ZEN was 98.0 % at 50 Micro gram g of the substrate. These findings differ from those reported by previous studies [32,39] who reported that the after 24 h purity declined to 89.0 %. The study of the biodistribution of \([^{125}\text{I}]\)-iodo ZEN revealed that \([^{125}\text{I}]\)-iodo ZEN rapidly accumulate in the blood, liver, intestine, and kidney. Such results follow the same pattern as the results of previous reports [39] and [40] who reported that the accumulation of \([^{125}\text{I}]\)-labeled deoxynivalenol in the liver and kidney was observed following intravenous administration in mice. Another report by [41] revealed the presence ZEN in the rats, ZEN was found in kidneys, liver, and lung after intravenous administration. According to the International Agency for Research on Cancer (IARC), zearalenone belongs to group No. 3, which includes mycotoxins that are not classified as carcinogenic to humans [42]. The toxicity of ZEN in vivo was studied by many research mostly reported the hazardous effect of Zearalenone on the reproductive systems either in male or female [42], also there was a specific concern about the lipid peroxidation of membranes which is included in the means by which ZEN exert its toxic action on human [413, another concern was the findings of [44] who demonstrated the ability of Zearalenone to disrupt liver functions and affect the liver thoroughly to the degree of suppression of cytochrome P450 in the liver results of our work gave clear data about the distribution of Zearalenone in different body organs which in turn could reveal the most susceptible organs that could suffer from Zearalenone/derivatives toxicity [121–145]. By examining the beneficial effect of L. plantarum on the organ exposure to Zearalenone/derivatives, results revealed that although the administration of L. plantarum did not significantly alter the accumulation of \([^{125}\text{I}]\)-iodo ZEN in different mice organs and there were observed similarity in the pattern of Zearalenone distribution through different organs yet there was marked reduction in the Zearalenone concentration in the ovary and that was observed from the 1st 5 min of administration, and since the ovaries are considered the most sensitive (target) organ of Zearalenone toxicity that could provide additional benefit for the use of LAB as it was reported to demonstrate anti-inflammatory, anti-genotoxic and improve ovarian fertility [45], the decline in the accumulation of \([^{125}\text{I}]\)-iodo ZEN in mice organs may be attributed to the lactic acid bacterial capability to bind and/or alter ZEN to other compound [9]. Which in turn could result in the decrement of the ZEN in the ovaries leading to the reduction in their related estrogenic
Fig. 4. (a) Variation of the radiochemical yield of radioiodinated ZEN as a function of different ZEN amounts; reaction conditions: 10 μl (~3.7 MBq) Na\(^{125}\)I, (x μg) olmesartan, 100 μg of Ch-T, at pH 6, the reaction mixtures were kept at room temperature for 30 min. (b) Graph showing the effect of different oxidizing agent amount on the radiochemical yields under the reaction conditions: 10 μl (~3.7 M Bq) Na\(^{125}\)I, 100 μg of ZEN, (x μg) of Ch-T, at pH 6, the reaction mixtures were kept at room temperature for 30 min. (c) Variation of the radiochemical yield of radioiodinated ZEN as a function of reaction time; reaction conditions: 10 μl (~3.7 MBq) Na\(^{125}\)I, 50 μg of ZEN, 100 μg of Ch-T, at pH 6, the reaction mixtures were kept at room temperature for different intervals of time. (d) Variation of the radiochemical yield of radioiodinated ZEN as a function of pH; reaction conditions: 10 μl (~3.7 MBq) Na\(^{125}\)I, 50 μg of ZEN, 100 μg of Ch-T, at different pH, the reaction mixtures were kept at room temperature for 30 min.
influence [46]. Established the ability of some lactic acid bacterial strains to remove ZEN and its derivatives from contaminated from liquid media.

4.1. Molecular modeling studies

The structural analysis for the Zearalenone ligand co-crystallized along with estrogen receptor alpha (ERα) (PDB code: 5krc) substrate-binding pocket revealed S-score of $-14.8831$ kcal/mol and hydrogen bonds: one with E353 of distance (1.34 Å), one with L346 of distance (1.34 Å), as shown in (Fig. 7).

The biodistribution studies in two groups of normal and infected mice labeling purity was observed to be stable in normal saline and rat serum. The biodistribution studies in two groups of normal and infected mice indicated that the radiotracer [125I]-ZEA has the same distribution except in ovaries. It was also observed that [125I]-ZEA accumulation declined in the ovaries at the group of mice taking L. Plantarum, infected group from 3.22 % to 0.09 at 2 h p.i. Therefore, this current 298 work proposes a possible attempt to diminish the GIT absorption of ZEA and/or other mycotoxins from the human diet.

For ZEN which is a widespread mycotoxin contaminating cereal crops, a lot of worries concerning the hazards effects resulting from acute ingestion or chronic ingestion of insanitary food as a result of

The results of the docking study were as follow:

Monoiodo-zearalenone suggested structure (2a) has S-score of $-13.5369$ kcal/mol and had two hydrogen bonds with 2 different amino acids: one with E353 of distance (1.35 Å), one with L346 of distance (1.34 Å), as shown in (Fig. 7).

Diiodo- zearalenone suggested structure (2b) has S-score of $-12.5866$ kcal/mol and had one hydrogen bond with H524 of distance (2.44 Å), as shown in (Fig. 7).

Monoiodo-zearalenone suggested structure (2c) has S-score of $-12.2297$ kcal/mol and had two hydrogen bonds with 2 different amino acids: one with E353 of distance (1.91 Å), one with H524 of distance (2.45 Å), as shown in (Fig. 7).

Conclusively, the presence of heavy iodine atom didn’t disturb the binding ability of zearalenone to its target (5krc).

5. Conclusions

An optimized protocol for the synthesis of the radiotracer [125I]-iodo ZEA in high radiochemical yield of ≥ 98 % has been developed by electrophilic substitution mediated by Ch-T. About 98 % radio-labeling purity was observed to be stable in normal saline and rat serum. The biodistribution studies in two groups of normal and infected mice indicated that the radiotracer [125I]-zearalenone has the same distribution except in ovaries. It was also observed that [125I]-iodo ZEA accumulation declined in the ovaries at the group of mice taking L. Plantarum, infected group from 3.22 % to 0.09 at 2 h p.i. Therefore, this current 298 work proposes a possible attempt to diminish the GIT absorption of ZEA and/or other mycotoxins from the human diet.

For ZEN which is a widespread mycotoxin contaminating cereal crops, a lot of worries concerning the hazards effects resulting from acute ingestion or chronic ingestion of insanitary food as a result of

| Table 1 | Biodistribution of $^{125}$I-Zearalenone in normal mice at different times. |
| --- | --- |
| Organs & body fluids | % LD$_g$/gram at different times post injection |
| | 5 min | 15 min | 30 min | 60 min | 120 min |
| Blood | 13.0 ± 0.16 | 10.2 ± 0.18 | 5.27 ± 0.18 | 2.0 ± 0.2 | 1.1 ± 0.1 |
| Bone | 1.14 ± 0.11 | 1.2 ± 0.1 | 1.11 ± 0.1 | 0.95 ± 0.02 | 0.90 ± 0.01 |
| Muscle | 3.6 ± 0.1 | 2.90 ± 0.2 | 2.00 ± 0.1 | 1.60 ± 0.1 | 1.00 ± 0.2 |
| Brain | 0.78 ± 0.01 | 0.70 ± 0.02 | 0.60 ± 0.01 | 0.55 ± 0.01 | 0.46 ± 0.01 |
| Lungs | 1.20 ± 0.11 | 1.30 ± 0.1 | 1.1 ± 0.15 | 0.91 ± 0.02 | 0.80 ± 0.01 |
| Heart | 1.00 ± 0.11 | 1.2 ± 0.12 | 0.91 ± 0.01 | 0.88 ± 0.02 | 0.80 ± 0.01 |
| Liver | 14.16 ± 0.17 | 18.60 ± 0.67 | 25.29 ± 0.25 | 38.24 ± 0.79 | 27.15 ± 1.10 |
| Kidneys | 5.17 ± 0.19 | 7.27 ± 0.99 | 10.25 ± 0.48 | 6.19 ± 0.89 | 3.11 ± 0.55 |
| Spleen | 1.11 ± 0.11 | 1.21 ± 0.14 | 1.25 ± 0.13 | 1.00 ± 0.11 | 0.89 ± 0.03 |
| Intestine | 12.4 ± 0.55 | 15.9 ± 0.66 | 29.14 ± 0.18 | 38.0 ± 0.78 | 45.8 ± 1.40 |
| Stomach | 1.11 ± 0.13 | 1.15 ± 0.12 | 1.0 ± 0.11 | 0.95 ± 0.01 | 0.90 ± 0.02 |
| Ovary | 2.50 ± 0.02 | 3.6 ± 0.03 | 8.8 ± 0.02 | 7.33 ± 0.01 | 3.22 ± 0.01 |

Mean±SEM (mean of five experiments).

| Table 2 | Biodistribution of $^{125}$I-Zearalenone in normal mice bearing acid lactic bacteria at different times. |
| --- | --- |
| Organs & body fluids | % LD$_g$/gram at different times post injection |
| | 5 min | 15 min | 30 min | 60 min | 120 min |
| Blood | 13.9 ± 0.17 | 11.1 ± 0.19 | 6.33 ± 0.17 | 3.0 ± 0.13 | 1.4 ± 0.11 |
| Bone | 1.15 ± 0.14 | 1.3 ± 0.09 | 1.00 ± 0.12 | 1.10 ± 0.13 | 0.98 ± 0.07 |
| Muscle | 3.1 ± 0.14 | 2.60 ± 0.16 | 2.21 ± 0.15 | 1.80 ± 0.12 | 1.10 ± 0.11 |
| Brain | 0.69 ± 0.02 | 0.72 ± 0.05 | 0.59 ± 0.06 | 0.50 ± 0.04 | 0.49 ± 0.01 |
| Lungs | 1.14 ± 0.11 | 1.20 ± 0.12 | 1.11 ± 0.13 | 0.95 ± 0.09 | 0.90 ± 0.08 |
| Heart | 1.12 ± 0.13 | 1.14 ± 0.12 | 1.00 ± 0.11 | 0.96 ± 0.1 | 0.91 ± 0.09 |
| Liver | 12.15 ± 0.19 | 17.77 ± 0.96 | 28.21 ± 1.87 | 36.20 ± 1.22 | 30.19 ± 0.22 |
| Kidneys | 4.16 ± 0.19 | 6.55 ± 0.33 | 9.21 ± 0.62 | 7.26 ± 0.27 | 2.98 ± 0.11 |
| Spleen | 1.17 ± 0.14 | 1.25 ± 0.15 | 1.20 ± 0.12 | 1.11 ± 0.11 | 1.00 ± 0.13 |
| Intestine | 11.5 ± 0.27 | 14.8 ± 0.59 | 30.15 ± 0.19 | 39.0 ± 0.55 | 46.1 ± 1.22 |
| Stomach | 1.00 ± 0.12 | 1.10 ± 0.13 | 0.95 ± 0.09 | 0.92 ± 0.08 | 0.90 ± 0.09 |
| Ovary | 0.11 ± 0.01 | 0.15 ± 0.02 | 0.22 ± 0.01 | 0.20 ± 0.03 | 0.09 ± 0.02 |

Mean±SEM (mean of five experiments).
mycotoxin contamination. This current work proposes a possible attempt to diminish the GIT absorption of ZEN and/or other mycotoxins from the human diet.

Authorship statement

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Conflict of interest

The authors declare no conflict of interest.

Data availability

Data will be made available on request.

Declaration of Competing Interest

The authors report no declarations of interest.

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Fig. 6. A) The violet circle is the main binding pocket of zearalenone co-crystalized ligand on (ERα) (PDB code: 5krc), B) Overlay of all the iodinated proposed structures and zearalenone on (ERα).

Fig. 7. 3D presentation of: zearalenone on (ERα) (PDB code: 5krc) showing the 2 H-bonds (dotted black line) with L387 and E353, show 2a with 2H- bonds (dotted black line) with H524 and E353 show 2b with one H- bond (dotted black line) with H524 and last 2c with 2 H-bonds (dotted black line) with L346 and E353.
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