Production of citric acid by *Aspergillus niger* using pineapple waste

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ABSTRACT

A solid state fermentation was developed for citric acid production from pineapple waste by *Aspergillus niger* KS-7. The medium was supplemented with different concentrations of glucose, sucrose, ammonium nitrate and ammonium phosphate. It was found that pineapple waste with 15% (w/v) sucrose and ammonium nitrate (0.25% w/v) gave the optimum citric acid secretion (60.61 g/kg) in the presence of methanol (2% v/v) when fermented for 5 days at 30 °C with the initial moisture content of 65%. The yield was more than 90% based on the amount of fermentable sugar consumed. These results present the use of pineapple peel as a cheap medium for the production of commercially valuable organic acid by *A. niger*.

Keywords: *Aspergillus niger*, citric acid, pineapple waste, solid state fermentation

INTRODUCTION

Citic acid worldwide demand is about 6.0 X 10^5 tons per year (Karaffa and Kubicek, 2003). Approximately 75% commercial use of citric acid is for food and 12% for pharmaceutical industries (Haq et al., 2001). Despite its wide application in food and pharmaceutical industries, citric acid is receiving little attention in the tropics.

Commercial production of citric acid is generally by submerged fermentation of sucrose or molasses using the filamentous fungus *A. niger* or synthetically from acetone or glycerol (Torres et al., 1998; Fernando et al., 2000; Adachi et al., 2003; Haq et al., 2004). In the recent times solid state fermentation (SSF) as an alternative to submerged fermentation in the production of microbial metabolites. Solid-state fermentations refer to the cultivation of microorganisms in a low-water-activity environment on a non-soluble materials acting as both nutrient source and physical support (Pandey, 2003). The major advantages of solid-state fermentation over submerged fermentation include higher yields, low water requirement and lower operating costs.

Many microorganisms have been evaluated for the production of citric acid including bacteria such as *Bacillus licheniformis*, *B. subtilis*, *Corynebacterium* spp. (Kapoor et al., 1983;); fungi such as *A. niger*, *A. awamori*, *A. foetidus*, *Penicillium restrictum* (Mattey and Allan, 1990); Kubicek, 1998). Yeast such as *Candida lipolytica*, *C. intermedia* and *Saccharomyces cerevisiae* (Crolla and Kennedy, 2001; Archer et al., 2001; Kamzolova et al., 2003). However, *A. niger* a filamentous fungus remained the organism of choice for citric acid production due to ease of handling, its ability to ferment a variety of cheap raw materials, and high yields (Schuster et al., 2002). A cost reduction in citric acid production can be achieved by using cheap agricultural wastes such as apple and grape pomace, orange peel, kiwi fruit peel, cotton waste, okara soy-residue and cane molasses (Kiel et al., 1981; Hang and Woodams, 1986; 1987; Khare et al., 1995; Haq et al., 2004).

Pineapple peel is a by-product resulting from the processing of pineapple into slices and represents about 10% w/w of the weight of the original fruit. Current disposal of it poses considerable economic and environmental problems. The objective of this study was to adopt the use of pineapple peel as a cheap medium for the production of citric acid by *A. niger*.

MATERIALS AND METHODS

Pre-treatment of pineapple peels

Peels from pineapples bought from a local market in Abeokuta Ogun state, Nigeria were used in the present study. Pineapple peels were oven-dried at 60 °C for 2 h and cut into 2 mm mesh size.

Screening of the fungal cultures

The *A. niger* cultures were screened qualitatively for the production of citric acid as described by Ali (2004). Czapek-Dox agar medium (10 mL) was poured into individual sterile Petri plates and allowed to cool at room temperature. Approximately 0.5 mL of the conidial suspension of *A. niger* was transferred to each of the Petri plates. The plates were incubated at 30 °C for 3–5 days. The plates were observed after incubation for yellow zones due to citric acid formation. Strains of *A. niger* with the widest yellow zone were used for further studies.

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Fermentation media

The basal medium was prepared by introducing pineapple peels (30 g) into 200 mL Erlenmeyer flasks. The medium was supplemented with glucose and sucrose at 5, 10, 15% w/v. Effect of nitrogen supplements was studied by adding ammonium nitrate (0.1–0.5%) and ammonium phosphate (0.1–0.5%) to the basal medium and moistened to varying moisture content (50–70%). The flask was cotton plugged and autoclaved at 121 °C for 15 min. After cooling at room temperature, each medium was inoculated with the A. niger (6.0 x 10⁶) suspension and incubated at different temperature range (25–30 °C) in a rotary shaking incubator for 5 days. Methanol (0–5%) was added to the flasks before fermentation. After fermentation, the medium was diluted with distilled water (1:4 w/v). The medium was then filtered and the filtrate used for the subsequent analyses.

Citric acid determination

Citric acid was determined titrimetrically (AOAC, 1995) by using 0.1 N NaOH and phenolphthalein as indicator and calculated as % according to the formula:

\[
\text{Normality} \times \text{volume of 0.1 M NaOH} \times \frac{\text{equivalent weight of citric acid}}{\text{dilution factor}} = \frac{\% \text{ citric acid}}{\text{Weight of sample (g) \times 10}}
\]

Biomass, residual sugars and pH determination

Biomass, sugar and pH values were determined according to AOAC (1995). To determine biomass the whole fungal culture was filtered with sterile filter paper and dried to a constant weight at 105 °C. Results were expressed in g/kg of pineapple peel. The sugar content was determined using a refractometer and pH was measured by Analog pH meter. Each analysis was conducted in triplicate.

RESULTS AND DISCUSSION

Effect of different concentration of sugars

Citric acid production by A. niger from pineapple peels as a basal fermentation media with the different concentrations of sucrose and glucose was shown in Table 1. The medium supplemented with sucrose (15% w/v) gave the highest citric acid value (36.6 g/kg) while the control (pineapple peels) gave 17.23 g/kg at 5 days fermentation period. Citric acid accumulation by A. niger in higher concentrations of sucrose or glucose is paralleled by a rise in the intracellular concentration of fructose 2.6 phosphate (Ali, 2004). Hossain and coworkers (1984) explained that the nature of sugar source has a marked effect on citric acid production by A. niger. In this study, addition of sucrose to pineapple waste enhanced citric acid production than glucose. Sucrose is the traditional commercial substrate for citric production although glucose, fructose and maltose have also been used as substrates for citric acid production (Xu et al., 1989). Sucrose is of relatively low molecular weight and is readily transported into microbial cells for hydrolysis by intracellular enzymes (Drysdale and McKay, 1995).

The increase in citric acid production and biomass values was accompanied with steady decrease in sugar along the incubation time. Kubicek (1998) reported that the final yield of citric acid in fermentation by A. niger is strongly dependent on the type and concentration of carbon source. Current understandings of mechanism by which the carbon source and it concentration influence citric acid accumulation were related to major regulatory points at the level of hexose transport and phosphorylation.

Effect of nitrogen supplements

Effect of nitrogen sources on citric acid productivity by A. niger was shown in Table 2. Supplementation of the basal medium with ammonium nitrate (0.25% w/v) gave an increase in citric acid production from 36.63 g/kg to 43.10 g/kg compared to ammonium phosphate (39.62 g/kg). Nitrogen had been reported to be an important factor in fermentation processes due to an increase in C/N ratio (Pandey, 2003). Any increase or decrease other than (0.25% w/v) concentration, resulted in the disturbance of fungal growth and subsequently citric acid production. However, addition of ammonium nitrate (0.5% w/v) gave a biomass value of 13.30 g/kg while optimum biomass (14.40 g/kg) was obtained at 1% w/v concentration of ammonium phosphate. Nitrogen constituent has a profound effect on citric acid production because nitrogen is not only important for metabolic rates in the cells but it is also basic part of cell proteins. This report agreed with Grewal and Kalra (1995) that fermentation media for citric acid biosynthesis should consist of substrates necessary for the growth of microorganism primarily the carbon, nitrogen and phosphorus sources.
Table 1: Effect of different sugar concentrations on citric acid production from pineapple waste by A. niger

| Fermentation media                  | Citric acid (g/kg) at different incubation time (days) |
|-------------------------------------|--------------------------------------------------------|
| Control                             | 0 | 1 | 2 | 3 | 4 | 5 |
| Pineapple waste                     | 0 | 13.2 | 13.6 | 14.7 | 16.6 | 17.2 |
| Pineapple waste + 5% glucose        | 0 | 14.3 | 15.7 | 16.8 | 19.8 | 19.7 |
| Pineapple waste + 10%glucose        | 0 | 16.1 | 18.6 | 22.8 | 26.2 | 26.2 |
| Pineapple waste + 15%glucose        | 0 | 16.2 | 18.8 | 24.2 | 26.5 | 29.4 |
| Pineapple waste + 5% sucrose        | 0 | 13.6 | 16.2 | 17.2 | 19.7 | 22.2 |
| Pineapple waste + 10% sucrose       | 0 | 17.3 | 19.1 | 21.4 | 25.8 | 29.7 |
| Pineapple waste + 15% sucrose       | 0 | 21.2 | 23.3 | 26.0 | 30.4 | 36.6 |

*Mean of triplicate determination SD±0.5

Table 2: Effect of Nitrogen supplements on citric acid production from pineapple waste medium

| Concentration (% w/v) | Citric acid (g/kg) | Biomass (g/kg) |
|-----------------------|--------------------|----------------|
| Control               | 36.63 ± 0.5        | 7.4 ± 0.2      |
| NH₄NO₃ 0.25           | 43.10 ± 0.8        | 10.2 ± 0.1     |
| 0.5                   | 38.20 ± 0.5        | 13.2 ± 0.2     |
| 0.75                  | 32.80 ± 0.6        | 8.2 ± 0.1      |
| 1.0                   | 26.80 ± 0.2        | 6.7 ± 0.2      |
| (NH₄)₂PO₄ 0.25        | 37.10 ± 0.5        | 9.79 ± 0.2     |
| 0.5                   | 39.62 ± 0.6        | 12.7 ± 0.2     |
| 0.75                  | 31.4 ± 0.5         | 14.2 ± 0.4     |
| 1.0                   | 26.2 ± 0.2         | 14.4 ± 0.3     |

Effect of incubation temperature

Effect of temperature on the production of citric acid was shown in Table 3. A temperature of 30 °C was found to be the best for citric acid production (43.5 g/kg) in the present study. The mold produced only a small amount of citric acid at 25 °C in five days. Sporulation however, was more marked at 35 °C than at lower temperatures. At low temperature, the low citric acid production was attributed to low enzyme activity. This report agreed with Hang and Woodams (1986) that the temperature of a fermentation medium is one of the critical factors that have a profound effect on the production of citric acid by solid state fermentation of agricultural wastes.

Effect of moisture content

Result shown in Table 4 indicated that a maximum citric acid value (51.6 g/kg) was obtained when the initial level of moisture was 65% which was 1.5 fold increases to value obtained at 55%. The importance of moisture level under SSF and its influence on the biosynthesis of microbial metabolites has been attributed to the interference of moisture in the physical properties of solid particles. Lower moisture level gives a lower degree of swelling and higher water tension, and then reduces the solubility of nutrients while higher moisture level decreases porosity, changes particle structure, promotes development of stickiness, reduces gas volume and exchange and decreases diffusion, which results in lower oxygen transfer (Lonsane et al., 1985)

Effect of methanol

Effect of methanol on citric acid production was shown in Table 5. Maximum citric acid production (60.6 g/kg) was obtained at 2% concentration. An increase in citric acid production at 2% methanol concentration was in agreement with Hossain et al. (1984) who stated that the presence of methanol in fermentation media may increase citric acid production by A. niger. The inductive effect of methanol for citric acid production may be due to reduction of the inhibitory effects of metal ions (Kiel et al., 1981). Moyer (1953) discovered the use of low molecular weight alcohols i.e. methanol, isopropanol as adjuncts to the culture medium which greatly increased citric acid production in both surface and submerged cultures. Such uses have made it possible to ferment directly crude carbohydrate substrates.

The citric acid, residual sugar, biomass and pH profiles

Citic acid production and residual sugar profiles during SSF of pineapple peel by A. niger are presented in Figure 1. Citric acid values steadily increased with fermentation time with a maximum of 60.61 g/kg at fifth day of fermentation. This result is quite comparable to the yields obtained by fermentation of other agricultural wastes such as kiwi fruit peel, soy-residue, and cane molasses (Hang
Table 4: Effect of moisture level on fungal production of citric acid from pineapple peel

| Moisture level (%) | Citric acid (g/kg) |
|-------------------|--------------------|
| 50                | 30.20 ± 0.8        |
| 55                | 34.20 ± 0.8        |
| 60                | 43.51 ± 1.1        |
| 65                | 51.11 ± 0.5        |
| 70                | 32.41 ± 0.5        |

Table 5: Effect of Methanol on citric acid production from Pineapple waste medium

| Concentration (% v/v) | Citric acid (g/kg) | Biomass (g/kg) |
|-----------------------|--------------------|----------------|
| Control               | 51.1 ± 1.0         | 10.2 ± 0.2     |
| 1                     | 53.4 ± 1.0         | 10.4 ± 0.2     |
| 2                     | 60.6 ± 1.4         | 8.2 ± 0.1      |
| 3                     | 56.2 ± 1.2         | 6.4 ± 0.1      |
| 4                     | 42.1 ± 0.8         | 5.1 ± 0.2      |
| 5                     | 29.5 ± 0.4         | 4.3 ± 0.2      |

Control: Pineapple waste supplemented with sucrose (15 g) and NH₄NO₃ (0.25 w/v) 2% methanol and incubated at 30 °C

In the present study, a parallel relationship between citric acid production and the consumption of sugar was also observed. This result agreed with the report of El-Holli and Al-Delamy (2003) that the production of citric acid approximately paralleled the consumption of sugar. At the end of the fermentation process a significant reduction in residual sugar from 68.1 g/kg to 4.5 g/kg was obtained. Hang and Woodams (1986) reported that the yields of citric acid from apple and grape pomace based on the amount of sugar consumed were about 88% and 60% respectively. In this work, based on the amount of fermentable sugar consumed, the yield of citric acid was more than 90% under optimum solid-state fermentation conditions.

Biomass is a fundamental parameter in the characterization of microbial growth. The most readily measured biomass component is protein (Raimbault, 1997). A steady increase in biomass throughout the fermentation period was observed with a maximum of 10.6 g/kg at fifth day while a decrease in pH from 4.0–2.5 was noted after 5 days of fermentation (Figure 2). The pH value maintained at the beginning of fermentation was important for a specific biomass formation. Normally, citric acid production occurred after 24 h of fermentation, this study shows that as incubation time increased more citric acid is produced and pH values decreased. Thus, the drop in pH observed during the process was due to the formation and accumulation of citric acid.

CONCLUSIONS

In conclusion, a solid state fermentation method has been developed for the production of citric acid from pineapple peel by A. niger. A maximum citric acid of 60.6 g/kg of pineapple peel was obtained under optimum conditions. This study indicates that the use of pineapple peel for fungal production of citric acid might represent an efficient method of minimizing pineapple waste disposal problems and concomitantly producing organic acids of valuable importance for food and pharmaceutical industries.

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