Critique of Primitive Model Electrolyte Theories

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Approximate theories for the restricted primitive model electrolyte are compared in the light of
Totsuji’s lower bound for the energy (an improvement over Onsager’s), Gillan’s upper bound for the
free energy, and thermal stability requirements. Theories based on the Debye-Hückel (DH) approach
and the mean spherical approximation (MSA), including extensions due to Bjerrum, Ebeling, Fisher
and Levin, and Stell, Zhou, and Yeh (PMSA1, 2, 3) are tested. In the range $T^* = k_B T D a / q^2 \lesssim$
$10 T_c^* \approx 0.5$, all DH-based theories satisfy Totsuji’s bound, while the MSA possesses a significant
region of violation. Both DH and MSA theories violate Gillan’s bound in the critical region and
below unless ion pairing and the consequent free-ion depletion are incorporated. However, the
PMSA theories, which recognize pairing but not depletion, fail to meet the bound. The inclusion of
excluded-volume terms has only small effects in this respect. Finally, all the pairing theories exhibit
negative constant-volume specific heats when $T^* \gtrsim 2 T_c^* \approx 0.1$; this is attributable to the treatment
of the association constant.

I. INTRODUCTION

The liquid-gas phase transition in electrolytes is of current interest because of puzzling experiments and
theoretical efforts to understand them. For recent reviews, see [1–3]. The primary model used is the restricted
primitive model (RPM) consisting of two oppositely charged, but otherwise identical, sets of $N_+ = N_-$ hard
spheres of diameter $a$ and charge per particle $\pm q$, immersed in a medium of dielectric constant $D$
(to represent the solvent) and volume $V$. We will restrict our attention to the RPM in $d = 3$ dimensions and use the reduced
temperature and density

$$T^* = k_B T D a / q^2 \quad \text{and} \quad \rho^* = a^3 \rho,$$  \hfill (1.1)

where $\rho = (N_+ + N_-) / V = N / V$; the Debye inverse screening length,

$$\kappa_D = (4 \pi q^2 \rho / D k_B T)^{1/2}, \quad \text{with} \quad x = \kappa_D a = \sqrt{(4 \pi q^2 / T^*)};$$  \hfill (1.2)

the reduced Helmholtz free energy density

$$\tilde{f}(\rho, T) = -F_N(V; T)/V k_B T;$$  \hfill (1.3)

and the reduced configurational energy per particle, $u$, defined via

$$N q^2 / D a \ u(\rho, T) = U_N - \frac{1}{2} N k_B T,$$  \hfill (1.4)

where $F_N$ and $U_N = V k_B T^2 (\partial \tilde{f} / \partial T)$ denote the total free energy and (internal) energy.

Recent theory [2, 3] has focussed on two approaches to approximating the free energy of the RPM, based on
either Debye-Hückel (DH) theory [3] or the mean spherical approximation (MSA) [1, 2]. Many years ago Bjerrum
[4] proposed to improve DH theory by including ion pairing via “chemical association.” Later, Ebeling
and Grigo [4] combined ion-pairing with an MSA expression for the ionic free energy; more recently, Levin
and Fisher [5] and Stell and coworkers [6] explored further extensions of the MSA. On the other hand, Fisher
and Levin [7] supplemented DH theory not only with ion pairing and excluded-volume terms but also included
the solvation free energy of the electrically active (+, −) dipolar ion pairs. Currently, this class of DH-based theories seems to give the best, albeit semiquantitative, account of the RPM in the critical region as judged by comparison with simulations performed by various authors [7, 8]. It may be remarked that the simulation estimates for $T_c^*$ and $\rho_c^*$ have been changing at an alarming rate [2(b)]. Nevertheless, the MSA-based theories yield approximations for $T_c^*$ ($\gtrsim 0.073$) that are significantly higher than those based on the DH approach ($T_c^* \lesssim 0.056$), which in fact agrees much better with the simulations ($T_c^* = 0.048$–0.056) [8, 9, 10].

At a purely theoretical level, however, one cannot be content since, a priori, there seem no clear grounds
for preferring the DH-based theories — apart from their more direct and intuitive physical interpretation —
rather than the more modern (and fashionable) MSA-based theories which — since they entail the pair correlation functions and the Ornstein-Zernike (OZ) relation — give the impression of being more firmly rooted in statistical mechanics. On the other hand, it has recently been shown that the DH theories yield pair correlations satisfying the OZ relation in a very natural way [11]. Furthermore, both theories have an essentially mean-field

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character despite which, in contrast to typical mean field theories for lattice systems, neither has any known Gibbs-Bogoliubov variational formulation or similar basis. How, then, might the two approaches be distinguished?

Now Blum and his coworkers have, in various places [17–20], enthusiastically sung the praises of the MSA for the RPM, asserting that the theory “is asymptotically correct in the limit of high density and infinite charge” or “high screening parameter (Debye length going to zero).” Furthermore, “unlike the DH theory, it [the MSA] satisfies the exact Onsager bounds for the Helmholtz free energy and the internal energy” [18] (in the same asymptotic limit) and the “internal energy of the MSA is an exact lower bound” [19]. As reported below, these claims cannot be sustained: however, they do suggest that one might usefully assess and compare the MSA and DH theories, and their various extensions, by checking their predictions against previously developed bounds for the internal energy and Helmholtz free energy. That task is undertaken here.

Indeed, as discussed more fully in Sec. II, several bounds have been established. The well-known Onsager lower bound for the configurational energy of the RPM was derived in 1939 [21]; less heralded is an improvement due to Totsuji some forty years later [22]. For the free energy, Rasahia and Stell [23] proved that the hard-core free energy provides an upper bound, while Gillan [24] developed a much stronger upper bound embodying the idea of (+, −) pairing into dipoles [25]. Finally, we note that thermodynamic stability with respect to temperature requires the positivity of the specific heat at constant volume [26].

We will focus particularly on the Totsuji and Gillan bounds applied in the region of the predicted gas-liquid phase transition and critical point. We find that DH theory and all its augmentations always satisfy Totsuji’s (and Onsager’s) bound provided $T^* \lesssim 10 T_c^* \simeq 0.5$. On the other hand, the MSA actually violates the Totsuji bound in a significant region of the $(\rho, T)$ plane where coexistence is predicted, unless the theory is suitably augmented.

In the light cast by Gillan’s bound, the two approaches rest on a more equal footing. As already shown by Gillan [24], the MSA (in its usual form) fails badly for $T^* \lesssim 0.08$; but the same is true for the original DH theory (even when supplemented by excluded-volume terms [24]). Only when both basic theories are augmented by ion-pairing contributions and by allowing for the associated depletion of the free-ion screening do they satisfy the Gillan bound. As against the hard-core electrostatic effects, included in both DH and MSA treatments, the presence or absence of specific excluded-volume terms has small effect numerically and does not affect the satisfaction of the bound. However, the recent PMSA (or pairing-MSA) theories of Stell and coworkers [3] violate Gillan’s bound apparently because they do not account appropriately for the free-ion depletion.

The main lesson is the crucial importance of the clustering of ions into dipolar pairs at low temperatures. Of course, this has been appreciated heuristically for a long time [13] and was quantitatively demonstrated in 1983 by Gillan [26] in calculations for the RPM which showed that the vapor for $T^* \lesssim 0.053$ consisted mainly of $(+, -), (2+, -), (3+, -), \ldots$ neutral clusters and $(2+, -), (1+, -), (3+, -), (2+, -)$ singly charged clusters, with relatively few free monopoles, $(+)$ and $(-)$. The present work, however, seems to be the first purely analytic demonstration of the thermodynamic necessity for including clustering, implicitly or, perhaps, explicitly, in approximate theories.

The recognition of $(+, -)$ ion-pairing requires the specification of the corresponding association constant, $K(T)$. Ever since Bjerrum’s original proposal [13], this has been a matter of confusion and contention (see, e.g., [2,4]). Nevertheless, in the low temperature region of principal interest here, say $T^* \lesssim 0.08 \simeq 1.5 T_c^*$. Bjerrum’s cutoff form and Ebeling’s more sophisticated expression agree to within 1.8% or better [2,3,6,4] and, along with other forms, have identical asymptotic expansions in powers of $T^*$ [2,7]. For practical purposes, therefore, $K(T)$ might be regarded as known “exactly.” At higher temperatures, where pairing should be (and is predicted to be) much weaker, it is natural to surmise that different treatments of association would prove inconsequential. However, this proves false! Indeed, for all the previous pairing theories [1,4,13] we find that the constant-volume configurational specific heat becomes negative (violating thermodynamics [25] and statistical mechanics) in the region $T^* = 0.1$ to 0.5: see Sec. IV. The source of this serious problem is found in the proposed behavior of the association constant. Initial steps towards amelioration are indicated, but the issue will be pursued in more detail elsewhere [28].

It should be mentioned that we also examine the generalized MSA (GMSA) [13] and variants of the MSA thermodynamics derived from the (approximate) pair correlation functions by routes other than the standard energy equation [1]: these are discussed in Sec. III. Other even less realistic models for electrolytes exist, including the one-component plasma with hard cores [28] and the corresponding “dense-point limit” [11(c)]; however, we address here only the RPM.

The explicit comparisons of the DH and MSA theories without allowance for ion pairing are presented in Sec. III, below. In Sec. IV the theories that include descriptions of ion pairing are assessed, including the PMSA theories [3].
II. BOUNDS FOR THE ENERGY AND FREE ENERGY

A. Configurational Energy Bounds

The first rigorous lower bound for the configurational energy of the RPM seems to be due to Onsager [21]. It is essentially a consequence of the positivity of the total electrostatic potential energy density and, with the notation of (1.4), yields

\[ u(\rho, T) \geq u_{\text{Ons}} = -1. \]  

(2.1)

A more transparent derivation for a system with a neutralizing background has been presented by Rosenfeld and Gelbart [30]. Totsuji, in 1981 [22], improved on Onsager’s result by writing the energy as an integral over the ionic pair correlation functions and showing that the presence of the hard-core repulsions implies an upper bound on the correlation functions. He thence established

\[ u(\rho, T) \geq u_{\text{Tot}} = -0.960. \]  

(2.2)

Although the improvement is by only 4.0%, it has significant consequences.

As remarked by Totsuji, one may usefully compare these bounds with the electrostatic or Madelung energies of an ionic crystal; for the NaCl (sc) and CsCl (bcc) structures one has [31]

\[ u_{\text{NaCl}} \simeq -0.8738 \quad \text{and} \quad u_{\text{CsCl}} \simeq -0.8813. \]  

(2.3)

One may reasonably suppose that the latter represents the best possible lower bound and so we will also invoke it in testing approximate theories for the RPM.

B. Gillan’s Free Energy Upper Bound

Gillan [24] has developed a convincing, but not fully rigorous, upper bound on the Helmholtz free energy of the RPM, which incorporates the idea of ion pairing. The pure hard-core free energy actually provides a rigorous upper bound [23], but Gillan’s bound is lower except for extremely low densities \((\rho^* \lesssim 10^{-5})\) where the limiting behavior is well understood. Here we utilize only Gillan’s bound, which is derived with the aid of the Gibbs-Bogoliubov inequality by employing a sequence of truncated reference systems. The calculation finally incorporates paired \((+, -)\) ions or dipoles by using a reference system of over-sized, spherically-capped cylinders with modified Coulomb interactions. The last step of Gillan’s argument relies on a comparison of an approximate analytical expression for the pressure of a system of such spherocylinders with computer simulation estimates [23]: the approximate formula appears to provide a bound on the true results. A search of the more recent literature concerning this system (e.g. Refs. [34–37]) indicates that the original simulations have withstood the test of time. (However, Frenkel [38] has observed that at high densities and for length/diameter ratios larger than needed here, the simulations — and, certainly, the analytic approximation — miss an isotropic-nematic fluid transition that is to be expected.) We thus believe that Gillan’s bound is valid.

To display the bound explicitly, we write the diameter and the chosen [24] center-to-center distance of the spherocylinders as \(a_s = (1 + \delta)a\) and put

\[ \lambda \equiv (5\pi/24)\rho a_s^3 = (5\pi/24)(1 + \delta)^3 \rho^*. \]  

(2.4)

If \(f^{1d}(\rho, T)\) is the ideal-gas free energy density, we then have [24]

\[ -f(\rho, T) \leq -f^{1d}(\rho, T) + \frac{1}{2} \rho \mathcal{F}(\rho, T), \]  

(2.5)

\[ \mathcal{F}(\rho, T) = 1 - 2\pi \rho - \frac{1}{T^*} \left[ \frac{18}{\pi} \lambda + \frac{1 - 2\lambda}{(1 - \lambda)^2} \right] - \ln \mathcal{L}(\rho, T), \]  

(2.6)

\[ \mathcal{L}(\rho, T) = T^*(1 - \lambda) \{ 1 - \exp[-\delta/T^*(1 + \delta)] \}. \]  

(2.7)

We will adopt \(\delta = 0.3\), which Gillan found optimized the bound for most values of \(T\).

III. BASIC THEORIES FOR THE RPM: COMPARISON WITH BOUNDS

A. DH and MSA without pairing

1. DH theory

Debye-Hückel theory [3] (here referred to as “pure” DH theory, since explicit dipolar pairing is not included) is the oldest theory for electrolytes still in current use. The theory entails two approximations: first, the pair correlation functions, \(g_{ij}(r_i - r_j)\), are represented by naive Boltzmann factors — with the charge, \(q_j\), multiplied by the average electrostatic potential at \(r_j\) when an ion of charge \(q_i\) is fixed at \(r_i\) — ignoring higher order correlation effects; and, second, these Boltzmann factors are linearized, which is valid only in the limit of low density, small charge, or high temperature. (For a modern discussion, see McQuarrie [4].) The thermodynamics predicted by DH theory depends only on the single parameter, \(x = \kappa_D a\). The appearance of the hard core diameter, \(a\), demonstrates that DH theory takes account of the electrostatic effects of the hard cores; however, the original or pure DH theory did not treat the excluded-volume effects of the hard cores (and so reduced to a theory for an ideal gas mixture in the limit of vanishing charge, \(q \to 0\)). Nonetheless, excluded volume contributions may be included naturally by adding to the free energy a suitably chosen pure hard-core term [18]; see below. In the DH critical region, such terms have a relatively small effect.
2. The MSA and variants

The other “basic” theory we consider, the mean spherical approximation \([10]\), is defined by a closure of the Ornstein-Zernike relation in which the \(g_{ij}(r)\) vanish inside the hard core, while the direct correlation functions outside the hard-core exclusion zone are approximated by the Coulombic potentials. Waisman and Lebowitz \([11]\) solved the MSA exactly for the RPM: that is, they determined the correlation functions which, in principle, yield the thermodynamics. The electrostatic free energy again depends only on \(x = \kappa_0 a\), but it and the overall free energy depend strongly on the theoretical route taken — via, in particular, the energy, pressure, or compressibility relations. Since very different results are obtained, we review them briefly. The standard MSA thermodynamics almost invariably discussed in the literature employs the energy route; but as a result, no excluded-volume hard-core terms are generated. Typically this problem is overcome by adding in appropriate terms “by hand,” just as for DH theory \([11]\). In light of this fact, the conceptual advantage sometimes claimed for the standard MSA in comparison to DH theory (see, e.g. \([8(b)]\)), namely, that the former treats the hard cores in better fashion, seems strictly inconsequential. Note also that the density-density correlation functions, \(G_{\rho\rho}(r)\), and also charge-charge correlation functions, \(G_{qq}(r)\), that satisfy the Stillinger-Lovett second-moment-condition follow from DH theory (again contrary to \([8(b)]\)) when properly generalized \([16]\).

The pressure route to MSA thermodynamics (which we will denote MSpA) generates a different approximation for the electrostatic excess free energy, along with the Percus-Yevick-pressure-equation hard-core free energy. It is interesting that, like the ordinary energy-route MSA thermodynamics, the MSpA yields both a critical point and the exact DH limiting laws; early on, however, Waisman and Lebowitz \([11(c)]\) dismissed it as inferior. By contrast, the compressibility route yields no electrostatic contribution, but generates only the Percus-Yevick-compressibility-equation free energy for uncharged hard spheres! Finally, note that the thermodynamics of the generalized MSA or GMSA (which is designed so that all three routes to the thermodynamics agree) \([14]\) is identical to the ordinary, energy-route MSA combined with the Carnahan-Starling (CS) approximation for the pure hard-core free energy \([33]\).

3. Hard Cores

Since the RPM consists of hard spheres, it is certainly desirable to include an account of the excluded volume effects in any approximate theory. As we have seen, the two principal approximations, DH and MSA, require the insertion of hard cores terms “by hand,” and two other theories, MSpA and GMSA, entail two different hard-core approximations. For the sake of convenience and uniformity, then, we will employ the CS hard core approximation \([33]\) in the calculations reported here for all theories that recognize excluded volume effects. The corresponding theories will be denoted DHCS, MSACS, and MSpACS, while the notation DH, MSA, and MSpA will be reserved for the “pure” (electrostatics only) theories. We have, however, checked that other approximations for the pure hard-core contributions yield qualitatively similar results.

It is worth mentioning that although hard-core terms do not contribute directly to the internal energy (since their contribution to the energy of allowed configurations vanishes — as correctly reflected by the CS approximation), they do influence the overall internal energy picture. Specifically, for the basic theories, as we shall see, they affect internal energy isotherms by altering the coexistence curve; for the augmented, pairing theories, they enter by changing the degree of pairing.

B. Assessment of Basic Theories

1. DH Configurational Energy

For pure DH theory (with neither pairing nor hard-core effects) the configurational energy assumes a particularly simple form, namely,

\[ u^{DH}(\rho^*, T^*) = -x/2(1 + x). \]  \(3.1\)

Evidently the energy of DH theory violates none of the bounds for any values of \(\rho\) and \(T\): see \([2, 21]\), and \([22]\). Furthermore, \(u^{DH}\) remains above the crystal values \([23]\) as is apparent in Fig. 1. The contrary statements by Blum and coworkers \([13, 19]\) that \(u^{DH}\) violates Onsager’s bound perhaps mistake the Debye-Hückel limiting law (DHL) — i.e., truncation of DH theory to lowest order in \(x\), which no one should take seriously for \(x \gtrsim 0.3\): see Fig. 1 — for the full DH theory propounded in \([1]\).

Strictly, the dependence of \(u^{DH}\) on the single parameter \(x\) given in \([2, 1]\) can be correct only in single-phase regions of the \((\rho, T)\) plane. Below the critical temperature (as defined by the theory at hand) the energy in the coexistence region is always a weighted sum of the values in the two phases, say \(\alpha\) and \(\beta\). In fact, if the energies per particle are \(u_\alpha\) and \(u_\beta\) and the densities \(\rho_\alpha = \rho_\alpha(T)\) and \(\rho_\beta = \rho_\beta(T)\), one finds

\[ u(\rho^*, T^*) = \frac{\rho_\alpha(\rho_\beta - \rho)u_\alpha + \rho_\beta(\rho - \rho_\alpha)u_\beta}{\rho(\rho_\beta - \rho_\alpha)}, \]  \(3.2\)

so that \(u\) varies linearly with \(1/\rho\). Thus the main DH plot in Fig. 1 is restricted to \(T \gtrsim T_{c}^{DH}\), and similarly for the other theories. However, including phase coexistence according to \([2, 2]\) cannot induce bound violation, since a weighted sum of two acceptable values also satisfies the
bound: see the inset in Fig. 1 where the solid curves depict DH isotherms for $T \leq T_c^{DH}$.

FIG. 1. The configurational energy per particle for the Debye-Hückel (DH), mean spherical approximation (MSA) and related theories above criticality for comparison, with lower bounds. For a description of the bounds and the theories, see the text. The inset shows isotherms for $T \leq T_c$ for the DH and DHCS theories as solid and dashed curves, respectively. (Here and below, CS denotes use of the Carnahan-Starling approximation for the excluded-volume effects.)

Regarding the effects of hard cores, one finds that the only changes in DHCS theory occur in the two-phase regions below $T_c^{DHCS}$: the energy isotherms are shifted from those of pure DH theory since the coexistence curve differs. The dashed curves in the inset to Fig. 1 show the rather small effects: the shifts mainly reflect the expected lowered densities on the liquid branch of the coexistence curve. Naturally, these changes cannot induce any violation of Totsuji’s bound or of the crystal limits.

2. MSA Configurational Energy

Now Blum and Bernard \[17,18\] have claimed the energy of the (pure) MSA, is “asymptotically correct.” However, as can be seen in Fig. 1, the MSA reduced excess energy, namely \[14\],

$$u^{MSA}(\rho^*, T^*) = -\left[1 + x - (1 + 2x)^{1/2}\right]/x,$$  \hspace{1cm} (3.3)

asymptotically approaches the Onsager bound of $-1$ but violates the Totsuji bound for $x \geq x_T \simeq 1200$ (as Totsuji noted originally \[22\]). Furthermore, $u^{MSA}$ lies below the crystal values for $x \geq x_X \simeq 125$.

In fact, even in the absence of Totsuji’s result, it is hard to make sense of the claim \[13,18\] that the MSA energy is asymptotically correct for the RPM in the limit of large $x$ by virtue of its approach to Onsager’s bound. Agreement with a bound is hardly proof of correctness \[14\]!

Furthermore, the limit $x \to \infty$ at fixed density implies $T^* \sim T/q^2 \to 0$; but at low temperatures, one expects crystalline phases to appear for $\rho^* \lesssim \rho_{max} = \sqrt{2}$ (for fcc sphere packing) \[2\] and these are not described by any of the theories under consideration.

It is worthwhile to interpret more explicitly the values $x_T$ and $x_X$, where violation by the pure MSA (no hard cores) occurs. On the liquid side of the coexistence curve, $x_T$ corresponds to violation when $T^* \leq 0.012 \simeq (0.14)/T_c^{MSA}$ and $x_X$ corresponds to $T^* \leq 0.035 \simeq (0.41)/T_c^{MSA}$. (The first violation temperature here is estimated with the aid of a low-temperature asymptotic analysis of the pure MSA coexistence curve \[12\] while the second follows directly from a numerical evaluation.) The solid curves in Fig. 2 demonstrate the effects.

FIG. 2. Comparison of the MSA energy with bounds for $T \leq T_c$ at multiples of $T^* = 0.01$ up to $T_c^* \simeq 0.0858$ (solid curves). The dashed curve shows the $T^* = 0.03$ isotherm for the GMSA for which, presumably, violations occur only at much lower temperatures.

The inclusion of hard-core terms (“by hand”) in the pure MSA changes the liquid-side coexistence curves more strongly than in DH theory. Thus for the MSA with CS terms or, equivalently, for the GMSA, the violations shift to much lower ratios of $T/T_c^{GMSA}$: this is clearly evidenced by the dashed coexistence isotherm shown in Fig. 2 for $T^* = 0.030 \simeq (0.38)T_c^{GMSA}$ (with $T_c^{GMSA} \simeq 0.0786$ \[13\]).

3. MSpA Configurational Energy

The energy according to the MSpA is \[11\]

$$u^{MSpA} = -\frac{1}{3} \left[1 - (1 - \sqrt{1 + 2x})/x + 2 \ln (1 + x + \sqrt{1 + 2x}) - 2 - \ln 4\right],$$  \hspace{1cm} (3.4)

which, in the single-phase region, also depends only on the parameter $x$. As evident from Fig. 1, however, this
violates the Totsuji and Onsager bounds at \( x_T \simeq 6.5 \) and \( x_0 \simeq 7.1 \), respectively. These results provide ample justification for a disparaging evaluation of the pressure-route thermodynamics for the MSA. For the remainder of this paper, we thus omit the MSpA.

### C. DH and MSA Free Energies

In the pure theories (in which Bjerrum ion pairing is not explicitly included) we find that both DH theory and the MSA violate Gillan’s free energy upper bound. The entire vapor branches of both coexistence curves, as well as both sides of the DH critical region, are in violation. As shown in Fig. 3, the violations remain when hard-core excluded volume corrections are included. The DHCS and GMSA treatments exhibit very similar features, for the low densities of interest. Note that in Fig. 3 we follow the coexistence prescription for the free energy corresponding to (3.3). Note also that non-violation on one branch of the coexistence curve (as on the GMSA liquid side) is at best a qualified virtue since the construction of the coexistence curve depends on the free energies on both sides. In light of these results it is clearly imperative to examine theories which allow for ion pairing.

### IV. ASSESSMENT OF ION-PAIRING THEORIES

#### A. Bjerrum and Beyond

To compensate for the effects of the DH linearization of the electrostatic Boltzmann factor, Bjerrum postulated association of “free” ions of (residual) density \( \rho_1 \) into “bound” neutral dipolar pairs of density \( \rho_2 \) so that the overall density is

\[
\rho = \rho_1 + 2\rho_2.
\]

In terms of the ideal-gas free energy density \( \bar{f}^{1d}(\rho_j, T) = \rho_j[1 - \ln(\Lambda_j^3 \rho_j/\zeta_j)] \) with mean thermal de Broglie wavelengths \( \Lambda_j(T) \) and internal partition functions \( \zeta_j(T) \), we may then write the total free energy density as

\[
\bar{f} = 2\bar{f}^{1d}(\rho_1) + \bar{f}^{1d}(\rho_2) + \bar{f}^{Ex}(\rho_1, \rho_2),
\]

with the excess free energy density

\[
\bar{f}^{Ex}(\rho_1, \rho_2) = \bar{f}^{HC}(\rho_1, \rho_2) + \bar{f}^{Ion}(\rho_1) + \bar{f}^{DI}(\rho_1, \rho_2),
\]

where (i) \( \bar{f}^{HC} \) denotes the pure hard-core/excluded-volume terms, (ii) \( \bar{f}^{Ion} \) represents the electrostatic contribution of the free ions, while (iii) \( \bar{f}^{DI} \) denotes the dipole-ion interaction/solvation terms. As mentioned, we take here \( \bar{f}^{HC} \) to be of Carnahan-Starling form with the dipoles treated as effective spheres of diameter \( \sigma_2 = 2^{1/3} \sigma \).}

#### FIG. 3. Comparison of the free energies predicted by the DHCS and GMSA theories in the density-temperature plane with Gillan’s upper bound. The bound is violated below the solid and dashed curves, respectively. For comparison, the associated coexistence curves with tie-lines and critical points are also plotted.

Chemical equilibrium among the + and – free ions and dipolar pairs is imposed via the equality \( \mu_2 = 2\mu_1 \) of the chemical potentials. If the association constant is defined by \( K(T) = A_1^3 A_2^3 \zeta_2/\zeta_1 - A_1^3 = \zeta_2 \) (see (3.3)) and the reduced excess chemical potentials are

\[
\bar{\mu}_j^{Ex} = \mu_j^{Ex}/k_BT = \ln \zeta_j = -(\partial \bar{f}^{Ex}/\partial \rho_j),
\]

with \( \rho_+ = \rho_- = \frac{1}{2}\rho_1 \) and \( \gamma_+ = \gamma_- = \gamma_1 \), then the mass action law states

\[
\frac{\rho_2}{\rho_+ \rho_-} = K(T; \rho_1, \rho_2) \equiv K(T)^{\gamma_+ \gamma_-}. \quad (4.5)
\]

The optimal expression for \( K(T) \) is a matter for debate — and will be discussed further below. For reference purposes we adopt Ebeling’s form which guarantees an exact representation of the RPM’s electrostatic second virial coefficient when one uses DH theory or the MSA (but not the MSpA) for \( \bar{f}^{Ion}(\rho_1) \). Note that for \( T^* \leq 0.05 \approx T_c^* \) the difference between \( K^{EB} \) and Bjerrum’s original proposal, \( K^{BJ} \), is less than 0.01%; it rises to 3.0% at \( T^* = T_c^{MSA} = 0.0858 \), in accord with the Introduction.

Bjerrum’s original theory amounts to the approximation

\[
\text{DHBj: } \bar{f}^{Ex} \simeq \bar{f}^{Ion} \simeq \bar{f}^{DH}(x_1) \quad \text{with } \quad x_1 = \kappa_1 \sigma. \quad (4.6)
\]

where \( \kappa_1^2 = 4\pi q^2 \rho_1/Dk_BT \) represents the inverse squared Debye length for the free ions alone, while as usual

\[
\bar{f}^{DH}(x) = \left[ \ln(1+x) - x + \frac{1}{2} x^2 \right]/4\pi \sigma^3. \quad (4.7)
\]
Friedman and Larsen [45] later found that the predicted coexistence curve was unphysical. More recently, Fisher and Levin [2,4,5] elucidated the peculiar “banana” shape of the DHBj coexistence curve (see Fig. 4 below) and showed it became worse when excluded-volume terms were added as, e.g., in DHBjCS theory. However, they also estimated the dipole-ion solvation term as [5]

\[
\bar{f}^{D1} = \rho_2(a a_2^2 / a_2^3 T^*) \tilde{\omega}_2(x_2), \quad x_2 = \kappa_1 a_2, \quad (4.8)
\]

where \(a_1 = (1.0-1.3)a\) is the mean dipolar size, or \(+/-\) ion separation, while \(a_2 \approx 1.1619a\) represents the effective electrostatic exclusion radius [5]. (Note that all the results given here use \(a_2 = a\) and \(x_2 = 1.1619a\).) The resulting DHBjDI theories lead to sensible coexistence curves (see Fig. 5 below) that agree fairly well with current simulations [5,2(b)].

At an earlier stage, Ebeling and Grigo [14] combined Bjerrum pairing with the MSA by replacing \(\bar{f}^{DH}\) by \(\bar{f}^{MSA}\)

\[
\bar{f}^{MSA}(x) = \left[ 2 + 6x + 3x^2 - 2(1 + 2x)^{3/2} \right] / 12\pi a^3, \quad (4.10)
\]

with \(x \Rightarrow x_1\) again evaluated at \(\rho_1\). They also added excluded-volume terms. The resulting MSABj and MSABjCS \(\equiv EGA\ [8(b),14]\) theories yield fully acceptable coexistence curves [5] but, as mentioned, the predicted critical temperatures are significantly too high [2(b),5].

Recently, Zhou, Yeh, and Stell (ZYS) [8] have extended Ebeling’s approach by using the MSA in conjunction with a “reference cavity theory of association” \[46\]. Their pairing mean-spherical approximations or PMSA theories may be described by

\[
\bar{f}^{E_x} = \bar{f}^{MSA}(\rho) + \rho_2(\rho, \rho) \ln(\gamma_1 + \gamma_2 / \gamma_2), \quad (4.11)
\]

where \(x = \kappa_D a\) is now evaluated with the total density, \(\rho\), and \(\bar{f}^{CS}\) represents the single-component Carnahan-Starling form, evaluated at \(\rho = \rho_1 + 2\rho_2\) (i.e., bound pairs are not treated as geometrically distinct objects). Note that \(\rho_2\) is here to be determined from \(\{4.5\}\) once \(K, \gamma_1\), and \(\gamma_2\) are specified (see below); hence \(\rho_2\) is an explicit algebraic function of the arguments stated in \(\{4.11\}\). The use of only the total density (in place of the free ion density \(\rho_1\)) results in an analytically simpler, more explicit formulation; but, in the light of the original DH and Bjerrum arguments, it seems rather unphysical since neutral bound pairs cannot contribute to screening in a direct way. Furthermore, as we will see, this approach entails a significant cost in accuracy.

The specification of the PMSA may be completed by first noting that ZYS also adopt Ebeling’s association constant, \(K^{E_b}(T)\) \[8(b),14\]. Then, for the activity coefficients, \(\gamma_+ \equiv \gamma_- \text{ and } \gamma_2,\) ZYS propose three levels of approximation, first:

\[
\text{PMSA1:} \quad \ln \gamma_1 = - (\partial \bar{f}^{MSA} / \partial \rho)_T \equiv \pi^{MSA}(T, \rho), \quad \gamma_2 = 1, \quad (4.12)
\]
which neglects dipole-ion contributions [cf. (4.11)]. Second, dipole-ion interactions are introduced by replacing the approximation $\gamma_2 = 1$ by

PMSA2:

$$\ln \gamma_2 = \frac{2(1 + x)\sqrt{1 + 2x - 2 - 4x - x^2}}{4T^* x^2}$$

$$\approx -\frac{x^2}{4T^* [1 + O(x)]}$$

(4.13)

see [8(b)], Eq. (4.11). Finally, the dumbbell-shaped hard cores of a dipolar ion pair are incorporated [8(a)] by using the CS cavity-value contact function and incrementing $\ln \gamma_2$ by

PMSA3: $\Delta \ln \gamma_2 = \ln \left[ \frac{2(1 - \eta)^3}{2 - \eta} \right]$, (4.14)

where $\eta = \pi \rho^*/6$.

PMSA3 is the preferred theory of ZYS and yields $(T_c^*, \rho_c^*) \simeq (0.0745, 0.0245)$. PMSA1 and PMSA2 give (0.0748, 0.0250) and (0.0733, 0.0229), respectively. The $T_c$ values are still significantly higher [8(b)] than the DH-based estimates, namely, $T_c^* \simeq 0.052-0.057$ [4(b),15], while the simulations suggest $T_c^* \simeq 0.048-0.055$ [2(b),15].

B. Pairing Theories vs. Gillan’s Bound

Comparison of the pairing theories with Gillan’s free energy bound is mainly encouraging. We find that theories that incorporate association in the Bjerrum chemical energy bound is mainly encouraging. We find that theoretical prescriptions, namely, $T_c^* \simeq 0.052-0.057$ [4(b),15], while the simulations suggest $T_c^* \simeq 0.048-0.055$ [2(b),15].

As regards the MSABj and DHBj theories, the more or less vertical “excess contour lines” in Fig. 4 reveal the magnitude of non-violation in the density-temperature plane: they are loci on which Gillan’s upper bound exceeds the corresponding approximate reduced free energy density, $-fa^3$, by the indicated amounts, ranging from $6 \times 10^{-4}$ up to 0.1. The associated coexistence curves are also shown and one may notice that the excess contours undergo a jump in curvature on entering the corresponding two-phase region: this results from the coexistence prescription analogous to (4.2).

Fig. 5 shows the effects of incorporating dipole-ion solvation (DI) and excluded-volume (CS) terms. Note that removing the excluded-volume terms from these BJDicS theories produces only slight shifts in the excess contours at high densities and low temperatures.

By contrast, the solid curve in Fig. 6 marks the boundary of the region inside which the PMSA3 free energy violates Gillan’s bound. The coexistence curve is also shown. (Note, however, that the coexistence prescription was not used here to compute the violation boundary within the two-phase domain.) The region of violation found for PMSA2 is nearly identical, while that for the PMSA1 theory is slightly larger, extending above the corresponding critical point, $T_c^{PMSA1}$; see the dashed curve in Fig. 6.

C. Pairing Theories vs. Energy Bounds

Testing the pairing theories against the bounds of Totsuji and Onsager yields mixed results. For a window of temperatures that includes the critical region, namely, $0.015 \lesssim T^* \lesssim 0.5$, all the theories embodying ion association satisfy the energy bounds. We also find a surprising level of agreement among the various theories as to the value of the critical energy per particle: see Table I. At low temperatures, however, some of the MSABj-based theories violate Totsuji’s bound. Moreover, at moderate
temperatures \((T^* \gtrsim 0.5)\) all of the pairing theories violate fundamental thermal stability requirements (as discussed in the next section); for some of the approximations, this is also accompanied by violation of the Totsuji and Ong-sager bounds, as explained below.

Now the energy for a general pairing theory follows from \(\frac{4}{2}\) via the thermodynamic relation \(\frac{1}{4}\) and the mass action law \(\frac{4}{3}\), etc., which leads to

\[
u(x, T) = a^3T^{x^2 / \rho^*} \frac{\partial f_{E^*}(x, T)}{\partial x} + \frac{\partial^2 f_{E^*}(x, T)}{\partial x^2} u_2(T), \tag{4.15}\]

where \(u_2(T)\) is given by

\[
u_2(T) = T^{x^2 / (d \ln K(T)/dT^*)} \tag{4.16}\]

But this can be recognized simply as the mean energy of a single \((+, -)\) bound pair since the corresponding internal configurational partition function for a pair is embodied in the association constant, \(K(T)\) — see the text above \(\frac{1}{4}\) and Levin and Fisher \(\frac{2}{3}\). Of course, the factor \(\rho_2(x, T)\) in \(\frac{4}{1}\) is also to be determined via the law of mass action \(\frac{4}{3}\). For theories of the form \(\frac{4}{3}\), one can further write

\[u^{E_{x}} = (a^3T^{x^2 / \rho^*})/(\partial f_{E^*}(x, T)/\partial x) = u_{\text{ion}} + u_{\text{DI}}, \tag{4.17}\]

where the “basic” expressions for the electrostatic contribution, \(u_{\text{ion}}\), are now given by the natural generalizations of \(\frac{3}{1}\) and \(\frac{3}{3}\), namely,

\[u^{DH}(x, T) = -(\rho_1 / \rho)x_1 / 2(1 + x_1), \tag{4.18}\]

\[u^{MSA}(x, T) = -(\rho_1 / \rho) \left[1 + x_1 - (1 + 2x_1)^{1/2}\right] / x_1. \tag{4.19}\]

For reference, we also quote the explicit result for \(u_{\text{DI}}\) following from the treatment of Fisher and Levin in leading order \(\frac{8}{8}\). Defining \(a_1\) and \(a_2\) as in \(\frac{8}{8}\) and \(\frac{8}{9}\), one finds

\[u^{DI} = -\frac{a_1^2 \rho_2}{2a_0 \rho} \frac{(\kappa a_2)^2}{3 + 3\kappa a_2 + (\kappa a_2)^2}. \tag{4.20}\]

The corresponding expressions for the PMSA theories are omitted for the sake of brevity.

### Table I. Some critical-point parameters for various theories: \(T^c\), \(u_c\), the reduced energy per particle; \(x_c\), the (overall) Debye parameter; and, \(x_1c\) = \((4\pi \rho_1c / T^c)^{1/2}\), the screening parameter. (Note that the values quoted for \(x_c\) in \(\frac{8}{8}\) correspond here to \(x_1c\), and that the Ebeling association constant \(\frac{1}{1}\) was used throughout.)

|            | DH   | +CS | +Bj  | +BjCS | +BjDI | +BjDI +CS |
|------------|------|-----|------|-------|-------|------------|
| \(T^c\)   | 0.0625 | 0.0615 | 0.0625 | 0.0615 | 0.0574 | 0.0525 |
| \(u_c\)   | -0.25  | -0.241 | -0.431 | -0.437 | -0.443 | -0.4535 |
| \(x_c\)   | 1     | 0.9315 | 3.0135 | 3.281 | 2.466 | 2.4240 |
| \(x_{1c}\) | 1     | 0.9315 | 1     | 0.9386 | 1.1229 | 0.9315 |

|            | MSA  | +CS | +Bj  | +BjCS | +BjDI | +BjDI +CS |
|------------|------|-----|------|-------|-------|------------|
| \(T^c\)   | 0.0858 | 0.0786 | 0.0858 | 0.0787 | 0.0821 | 0.0723 |
| \(u_c\)   | -0.4142 | -0.3358 | -0.4157 | -0.3781 | -0.4442 | -0.4148 |
| \(x_c\)   | 2.4142 | 1.522 | 2.7213 | 2.0408 | 3.0729 | 2.2083 |
| \(x_{1c}\) | 2.4142 | 1.522 | 2.4142 | 1.5319 | 2.4509 | 1.4850 |

|            | PMSA1 | PMSA2 | PMSA3 |
|------------|-------|-------|-------|
| \(T^c\)   | 0.0733 | 0.0748 | 0.0745 |
| \(u_c\)   | -0.3748 | -0.426 | -0.4265 |
| \(x_c\)   | 1.9814 | 2.0494 | 2.0329 |

### 1. Low Temperatures: Violation in MSA Pairing Theories

For \(T^* \lesssim 0.015\), evaluation of \(u(x, T)\) reveals violations of the Totsuji bound for most of the MSA theories. The reason turns out to be literally the same as for the pure MSA: in the corresponding Bj, BjCS, BjDI, and BjDICs theories, as well as in the PMSA1 (although not PMSA2 and 3) theory, the mass-action pairing predicted by \(\frac{1}{1}\) becomes exponentially small as \(T^* \to 0\) \(\frac{19}{19}\). As a result, all these theories revert to their ion-only form (i.e., MSA or MSACS) and violations occur: see Fig. 1. A similar depletion of pairs occurs when \(T^* \to 0\) in the DHBDI/CS (but not DHBDI/CS) theories, and so these theories revert to the corresponding non-violating DH/CS theories. These results are independent of whether one uses the Ebeling or Bjerrum association constant, or any other reasonable partition-function-like form, as discussed below.
2. Moderately Low Temperatures

In the temperature range $0.015 \lesssim T^* \lesssim 0.5$, which includes $T_e^*$, all the pairing theories described in the present study satisfy the Totsuji bound, and hence Onsager’s as well. Fig. 7 depicts energy isotherms for the pairing theories at $T^* = 0.07$. The plotted isotherms have been cut off for large $x = \kappa_D a$ at the hard-core packing limit, $\rho_{\text{max}} = \sqrt{2}$. Fig. 7 also shows the location of the critical point of the DHBJDICS theory, which may be regarded as a reference point in reading Table I. The table lists the various critical energies and Debye parameters. As mentioned, there is a fair measure of agreement among the different pairing theories regarding the energy at critically even though other parameters vary quite strongly.

![Fig. 7](image)

FIG. 7. Plots of configurational energy isotherms for various theories at $T^* = 0.07$, which temperature lies above all DH-based estimates of $T_e^*$ but below all MSA-based values. The scalled sections of the latter isotherms thus represent the two-phase regions. As regards the theories, recall that GMSA is equivalent to MSACS and note that the “+” implies the BjDICs extensions of the basic theories. At large $x = \kappa_D a$ all isotherms have been cut off at the hard-sphere close packing density. For reference, the critical point of the DHBJDICS theory (where $T_e^* = 0.0525$) has been marked.

3. Violations at Moderate and High Temperatures

Violation of the energy bounds are found again, as mentioned above, at higher temperatures in the range $T^* \gtrsim 0.5$, some 6 to 10 times greater than the estimates for $T_e^*$. The reason for this surprising fact, however, is quite different from the cause at low temperatures: it transpires, indeed, that the form of the association constant is now crucially important.

In fact, any theory with pairing governed by Bjerrum’s association constant violates both the Totsuji and Onsager bounds when $T^* \to \frac{1}{2}$ — and $\rho$ is large enough! Once noticed numerically, this behavior can be understood analytically by evaluating the factor $u_2(T)$ in (4.13) using (1.10) with $K = K_B^J(T)$. To that end recall, first, the well known fact [5] that $K_B^J(T)$ vanishes linearly, say as $c_B^J(1 - 2T^*)$, when $T^* \to \frac{1}{2}$ — and remains identically zero for $T^* > \frac{1}{2}$. Consequently, $u_2(T)$ diverges to $-\infty$ like $-\frac{1}{2}(1 - 2T^*)$ in this limit. However, the factor $\rho_2(\rho, T)$ in (1.13) must be evaluated via the mass action law (4.5) and is proportional to $K_B^J(T)$; this gives

$$\frac{\rho_2}{\rho} u_2 = \frac{\rho_*^2 \gamma_2 T^*}{4 \rho \gamma_2} \frac{dK}{dT^*} \approx \frac{c_B^J \gamma_2^2 \rho^*}{8 a^3 \gamma_2} < 0, \quad (4.21)$$

as $T^* \to \frac{1}{2}$, so that $\rho_2 \to 0$ and $\rho_1 \to \rho$. Note that the $\gamma_i(\rho, T)$, defined via (4.4), depend on the theory under consideration. One finds that $c_B^J/8a^3 \approx 11.6$: this is large enough so that the pairing term (4.21) by itself yields a violation of Onsager’s bound when (in DHBJ) $\rho^* > \rho_{\text{oms}}^*$ or (for MSABJ) $\rho^* > \rho_{\text{oms}}^* \approx 0.64$. However, as the other terms in (1.15) are also negative, violations must arise at even lower densities. One finds numerically, in fact, that the violations occur at or below $\rho^* \lesssim 0.3$ in all the theories with pairing governed by Bjerrum’s association constant.

One expects Ebeling’s choice, $K_{E_b}^J(T)$, which provides a match to the exact RPM second virial coefficient and never vanishes [5,14,44] — in contrast to the singular specific heat is a thermodynamic necessity dictated by the singular association constant violations at Moderate and High Temperatures

D. Violations of Thermal Stability

To pursue further the origins of the Totsuji and Onsager energy bound violations at $T^* \gtrsim 0.5$, consider the energy isochores shown in Fig. 8. The two densities $\rho^* = 0.03$ and 0.1 have been chosen for display because they bracket the critical density; similar behavior is seen at higher and lower densities. For the pure DH and MSA theories, included in Fig. 8 for reference purposes, $u(\rho, T)$ rises monotonically with $T$: this implies a positive constant-volume configurational specific heat, $C_V^{\text{conf}}(\rho, T)$. (Note that outside the two-phase region these two energy isochores are identical to those for DHCS and GMSA, respectively.)

Now the positivity of the total constant-volume specific heat is a thermodynamic necessity dictated by the
Second Law [25]. For a classical particle system, however, the configurational contribution must be separately nonnegative: this follows either, thermodynamically, by regarding the kinetic and configurational degrees of freedom as thermally distinct systems or, from statistical mechanics, by expressing $C_V^{\text{conf}}(\rho, T)$ as a mean-square energy fluctuation which is necessarily positive at finite positive temperatures in any nontrivial system [24].

![Energy isochores](image)

**FIG. 8.** Energy isochores at $\rho^* = 0.03$ and 0.1 (note shifted vertical scales) for the basic DH and MSA theories and for various pairing theories — solid lines for those based on DH, dashed lines for MSA-based. Ebeling’s association constant is employed for all plots excepting the four bracketed isochor es — the solid and dashed curves for DH-based, MSA-based, respectively, $u^{\text{DICS}}$ and $u^{\text{MSA}}$ into the two-phase regions loses all significance. [Note that, for the approximations considered here, $u^{\text{DICS}}(\rho, T)$, $u^{\text{MSA}}(\rho, T)$ outside the respective two-phase regions (see Sec. IIIA.3).]

However, a quick perusal of Fig. 8 shows that all the pairing theory isochor es — the solid and dashed curves representing DH- and MSA-based theories, respectively, and the dot-dash plots for PMSA1 and 3 — display regions where $u(\rho, T)$ decreases as $T$ increases. In other words, all the pairing theories predict negative constant-volume specific heats and violate the Second Law!

The reason is not far to seek. In the limit of complete pairing (i.e., $\rho_1 = 0$, $\rho_2 = 2\rho$) all the approximation theories under consideration predict, via (1.13), that the energy should be simply that of independent dipolar pairs: this corresponds to the plots labeled $\frac{1}{2}u_{B^1}^{B^1}$ and $\frac{1}{2}u_{B^2}^{B^2}$ in Fig. 8 which derive from (1.10) and the Bjerrum and Ebeling forms for $K(T)$. But, as evident from the figure, both $u_{B^1}^{B^1}(T)$ and $u_{B^2}^{B^2}(T)$ exhibit pronounced maxima in the interval $T^* = 0.12 - 0.13$ and then fall sharply as $T$ increases, dropping below $u_{B^1}^{B^1}(0) = u_{B^2}^{B^2}(0) = -1$ at $T^* = 0.222$ and 0.219, respectively. It is this behavior that leads to the decreasing regions in the overall excess energy isochor es with incomplete pairing. But such a variation of $u_2(T)$ is physically nonsensical since, clearly, the configurational energy $\varepsilon_2(r) = -q^2/Dr$ of a bound pair cannot fall below the contact value $-q^2/Da$ (which, in turn, can be achieved in equilibrium only at $T = 0$).

The problem with $u_2(T)$ arises because the defining relation (1.14) does not actually yield the physically anticipated thermodynamic mean value [24], say $\langle \varepsilon_2(r) \rangle_K$, which in the Bjerrum picture of association would be

$$\langle \varepsilon_2(r) \rangle_K = 4\pi \int_a^R \varepsilon_2(r) e^{-\beta \varepsilon_2(r)} r^2 dr / K(T),$$

with association constant

$$K(T) = 4\pi \int_a^R e^{-\beta \varepsilon_2(r)} r^2 dr.$$

The reason for the failure is simple: the Bjerrum cutoff $R$ is taken to be temperature dependent [51], explicitly, $R_{B^1}(T) = a/2T^*$ for $T^* = \frac{1}{2}$ [3]. In general, such temperature dependence leads to the difference

$$\frac{q^2}{Da} u_2(T) - \langle \varepsilon_2 \rangle_K = \frac{4\pi R^2 e^{-a/T^*} R}{K(T)} k_B T^* \frac{dR}{dT},$$

which is negative whenever $R(T)$ decreases as $T$ rises and which diverges when $K(T) \to 0$. The Ebeling association constant can also be written in the form (1.23) but with the large-$T$ asymptotic form $R_{E^2}(T) = a \approx a/12T^{*4}$ [51], which is quite accurate once $T^* \gtrsim 0.3$. We must conclude that neither the Ebeling nor the Bjerrum association constants can be regarded as representing even an “effective” partition function for an isolated ion pair as is required by or implicitly assumed in the standard theories of association [3].

As suggested by Fig. 8, the unphysically large values of $u_2(T)$ lead to negative specific heats over large regions of the $(\rho, T)$ plane when either $K_{B^1}(T)$ or $K_{B^2}(T)$ is employed. Fortunately for our primary focus on the critical region, the violations of thermal stability are confined in all cases to $T^* \geq 0.12 \geq 2T^*$ (and for the PMSA theories to $T^* \gtrsim 0.35$). At densities below $\rho^* = 0.01 - 0.02 < 0.6 \rho^*_c$ the pairing is sufficiently weak that the predicted $C_V^{\text{conf}}(\rho, T)$ always remains positive — although it does display an unphysical oscillation. Once violations arise at a given $T$, moreover, they persist to the highest densities.

Of course, certain features are specific to the choice of association constant. As remarked earlier, $K_{B^1}(T)$ “switches off” abruptly at $T^* = \frac{1}{2}$ where a nonphysical latent heat is implied for all $\rho > 0$; above $T^* = \frac{1}{2}$ pairing
is lost and no violations remain. When $K^{\text{EB}}(T)$ is used in DH- and MSA-based theories with excluded-volume terms, violations remain at the highest temperatures.

What might be a cure for these pathologies? It is clear from \((4.22)-(4.23)\) that the unphysical behavior of $u_2(T)$ can be avoided if one fixes the cutoff in \((4.23)\), say $R = a$, so defining $K^\lambda(T)$. Furthermore, for any fixed $\lambda > 1$, the low-$T$ behavior of $K^\lambda(T)$ still matches $K^{\text{EB}}(T)$ to all orders in $T^*$ in \((4.24)\). In addition, the choice of $\lambda$ may be optimized by requiring that the deviation \(|(K^{\text{EB}}/K^\lambda) - 1| \equiv \delta\) remain less than a specified level up to as high a temperature as possible. Thus one finds that $\lambda \simeq 3.4$ provides 1% precision ($\delta = 0.01$) up to $T^* \simeq 0.11$.

One can then check that none of the pairing theories employing $K^\lambda(T)$ with $\lambda \simeq 3.4$ violates the energy bounds or thermal stability for any realizable thermodynamic state, $(\rho, T)$. In addition, the qualitative conclusions regarding the violation and nonviolation of the Gillan free energy bound remain unchanged. Indeed, using $K^{3,4}(T)$ causes only insignificant shifts of the free energy excess contours from those displayed in Figs. 4 and 5 when $T^* \lesssim 0.1$.

Nevertheless, merely replacing $K^{\text{EB}}(T)$ by $K^\lambda(T)$ leads to significant inaccuracies in the thermodynamics at higher temperatures. Thus a more thoughtful approach is essential to providing a reasonable approximate theory of the RPM that is valid over the full range of temperatures and up to moderate densities excluding, of course, the solid phase(s)]. Such a treatment will be presented elsewhere \[28\].

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[1] J.M.H. Leveit Sengers and J.A. Given, Molec. Phys. **80**, 989 (1993).
[2] (a) M.E. Fisher, J. Stat. Phys. **75**, 1 (1994); (b) J. Phys. Cond. Matt. **8**, 1 (1996).
[3] G. Stell, J. Stat. Phys. **78**, 197 (1994).
[4] M.E. Fisher and Y. Levin, Phys. Rev. Lett. **71**, 3826 (1993).
[5] Y. Levin and M.E. Fisher, Physica A **225**, 164 (1996).

The following corrections should be noted: (a) In Eq. (4.16) the coefficient $-8$ in the correction factor should be: $-4$. (b) In Eq. (4.21) the last term should be: $-\frac{1}{3}u_{5m}^3[1 + O(\ell^2)],$ i.e., insert a factor 1/5. (c) In Eq. (5.10) the last factor $(b - 4)$ should be preceded by a solidus: “/”. (d) The line before Eq. (7.12) should read: “... charging process (3.7).” (e) The line in Eq. (9.2) should read: $2.141213 \cdots \cdots$. (f) The first member of Eqs. (9.6) should read $\rho_{\ast \ast} / \rho_0 \simeq 32T^{-3}$. (g) In the first member of Eqs. (9.7) the factor $T^{-2}$ should read $T^{-3}$. (h) See also \[2\] below.

[6] B. Guillot and Y. Guissani, Molec. Phys. **87**, 37 (1996).
[7] R.J.F. Leote de Carvalho and R. Evans, Molec. Phys. **83**, 619 (1994).
[8] (a) Y. Zhou, S. Yeh, and G. Stell, J. Chem. Phys. **102**, 5785 (1995); (b) S. Yeh, Y. Zhou, and G. Stell, J. Phys. Chem. **100**, 1415 (1996); and references therein.
[9] P.W. Debye and E. Hückel, Phys. Z. **24**, 185 (1923). For a modern account, see D. A. McQuarrie, Statistical Mechanics (Harper Collins, New York, 1976), Chap. 15.
[10] J.L. Lebowitz and J. Percus, Phys. Rev. **144**, 251 (1966).
[11] E. Waisman and J. L. Lebowitz, (a) J. Chem. Phys. **52**, 4307 (1970); (b) ibid **56**, 3086 (1972); and (c) ibid **56**, 3093 (1972).
[12] J.S. Hoye, J. L. Lebowitz, and G. Stell, J. Chem. Phys. **61**, 3253 (1974).
[13] N. Bjerrum, Kgl. Dan. Vidensk. Selsk. Mat.-fys. Medd. **7.1** (1926).
[14] W. Ebeling and M. Grigo, Ann. Phys. (Leipzig) **37**, 21 (1980); M. Grigo and W. Ebeling, J. Soln. Chem. **13**, 321 (1984); but note that these works suffer from numerical and other deficiencies: see, e.g., Ref. 8(b).
[15] G. Orkoulas and A.Z. Panagiotopoulos, J. Chem. Phys. **101**, 1452 (1994); J.M. Caillol, D. Levesque, and J.J. Weis, Phys. Rev. Lett. **77**, 4039 (1996); A.Z. Panagiotopoulos, private communication.
[16] B.P. Lee and M.E. Fisher, Phys. Rev. Lett. **76**, 2906 (1996) and to be published.
[17] L. Blum and O. Bernard, J. Stat. Phys. **79**, 569 (1995).
[18] O. Bernard and L. Blum J. Chem. Phys. **104**, 4746 (1996).
[19] L. Blum, M.F. Holvoko, and I.A. Protsykevych, J. Stat. Phys. **84**, 191 (1996).
[20] J.-P. Simonin and L. Blum, J. Chem. Soc. Faraday Trans. **92**, 1533 (1996).
[21] L. Onsager, J. Phys. Chem. **43**, 189 (1939).
[22] H. Totsuji, Phys. Rev. A **24**, 1077 (1981).
[23] J. Rasaiah and G. Stell, Molec. Phys. **18**, 249 (1970).
[24] M.J. Gillan, Molec. Phys. **41**, 75 (1980).
[25] (a) A.B. Pippard, Elements of Classical Thermodynamics for Advanced Students of Physics (Cambridge Univ. Press, 1957) Chap. 7; (b) H.B. Callen, Thermodynamics and an Introduction to Thermostatistics, 2nd Edn. (Wiley and Sons, New York, 1985) Chap. 8.
[26] M.J. Gillan, Molec. Phys. **49**, 421 (1983).
[27] An association constant with fixed upper cutoff was considered by R.M. Fuoss, Trans. Faraday Soc. **30**, 967 (1934), who also recognized the low-temperature insensitivity of $K(T)$ of the form $b/4T$ to the upper cutoff.
[28] D.M. Zuckerman and M.E. Fisher, to be published.
[29] R.G. Palmer and J.D. Weeks, J. Chem. Phys. **58**, 4171
(1973).

[30] Y. Rosenfeld and W. M. Gelbart, *J. Chem Phys.* **81**, 4574 (1984).

[31] C. Kittel, *Introduction to Solid State Physics*, 3rd Edn. (Wiley and Sons, New York, 1966).

[32] I. Nezbeda and T. Boublík, *Czech. J. Phys. B* **28**, 353 (1978).

[33] D.W. Rebertus and K.M. Sando, *J. Chem. Phys.* **67**, 2585 (1977).

[34] M.P. Allen, *Phil. Trans. R. Soc. Lond. A* **344**, 323 (1993).

[35] H.-L. Vörtler and J. Heybey, *Molec. Phys.* **51**, 73 (1984).

[36] D. Levesque, J.J. Weis, and J.P. Hansen, in: *Monte Carlo Methods in Statistical Physics*, ed. K. Binder (Springer, Berlin, 1979).

[37] P.A. Monson and M. Rigby, *Chem. Phys. Lett.* **58**, 122 (1978).

[38] D. Frenkel, *J. Phys. Chem.* **91**, 4912 (1987). See also D. Levesque and J.J. Weis in *The Monte Carlo Method in Condensed Matter Physics*, 2nd. Edn., ed. K. Binder (Springer, Berlin, 1995).

[39] N.F. Carnahan and K.E. Starling, *J. Chem. Phys.* **51**, 635 (1969); for mixtures, see (b) G. A. Mansoori et al., *J. Chem. Phys.* **54**, 1523 (1971); (c) G. Jackson, *Molec. Phys.* **72**, 1365 (1991); (d) for the corresponding Helmholtz free energy, see also Ref. [40].

[40] See Eq. (5) in Ref. [11(a)].

[41] For completeness we mention that Refs. [17] and [18] also cite the HNC approximation and compare the MSA favorably to it, in particular, for the one-component plasma (OCP), which consists of point particles. For the OCP, E.H. Lieb and H. Narnhofer, *J. Stat. Phys.* **12**, 291 (1975), have established a lower bound on the energy.

In the strong coupling limit, \( q^2 \rho^{1/3} / T \to \infty \), it has been shown by H. Gould, R.G. Palmer, and G.A. Estévez, *J. Stat. Phys.* **21**, 55 (1979), that the OCP energy computed via the MSA approaches this bound asymptotically; the same seems true for the HNC approximation: see K.-C. Ng, *J. Chem. Phys.* **61**, 2680 (1974). However, the relevance of these observations to the reliability or accuracy of the HNC and MSA for the OCP model again seems obscure.

[42] See Eqs. (9.8) and (9.9) of Ref. [5] but note the following corrections: (a) In Eq. (9.8) the coefficient \(-6\sqrt{2} \) in the correction factor should be: \(-18\); (b) In Eq. (9.9) the right-hand side should read: \( (e^3 / 81 \pi T^* e^{-1/T^*}) [1 - 92/3 T^* + \ldots] \).

[43] M.M. Telo da Gama, R. Evans, and T.J. Sluckin, *Molec. Phys.* **41**, 1355 (1980).

[44] W. Ebeling, *Z. Phys. Chem.* (Leipzig) **238**, 400 (1968). For explicit expressions for \( K_{Eb}(T) \), see [4] Eqs. (6.5)-(6.7).

[45] H.L. Friedmann and B. Larsen, *J. Chem. Phys.* **70**, 92 (1979).

[46] Y. Zhou and G. Stell, *J. Chem. Phys.* **96**, 1504, 1507 (1992).

[47] M.E. Fisher and B.P. Lee, *Phys. Rev. Lett.* **77**, 3561 (1996).

[48] See Eqs. (7.12) and (7.13) of [5].

[49] See Eqs. (9.15) - (9.18) of [5].

[50] See, e.g., in Ref. [25(b)], Sec. 19-2.

[51] O. Halpern, *J. Chem. Phys.* **2**, 85 (1934).

[52] N. Davidson, *Statistical Mechanics* (McGraw-Hill, New York, 1962), Chap. 7.