1. Introduction

Environmental pollution is accelerating because of an increase in various wastes generated by industrial development towards the improvement of living standards. Accordingly, the amount of plastics generated as waste is also increasing, and this has led to research attention being focused on developing a method of treating waste plastics. Existing landfill facilities have reached their limit, and therefore, there is a growing social concern regarding the shortage of landfill facilities and environmental pollution treatment. Thus, there is a need to develop environment-friendly technologies that can convert waste plastics into energy resources. In this study, we designed and fabricated a complex system via a pilot scale thermal plasma for solid fuel recycling of plastics through gasification and combustion to process large-capacity solid refuse fuel (SRF), and to study reactor temperature characteristics and generated gases [1-5].

Direct incineration is a method of directly oxidizing organic materials by supplying a large amount of air; waste heat and various oxidizing pollutants are generated as by-products. In the pyrolysis gasification melting method, organic materials are synthesized in a reducing atmosphere and inorganic materials are melted to produce slack and recycled. Table I summarizes the differences between the two systems.

The characteristics of the pyrolysis gasification melting technology are as follows: 1) C and H components in the waste are converted to CO and H2, thereby minimizing energy loss due to cooling; 2) The material decomposition process is performed in a reducing atmosphere under a low oxygen state, and the total amount of gas to be treated is small and no harmful substances are generated. The generated substances are decomposed during the high temperature gasification reaction, which is economically and environmentally advantageous over incineration; and 3) Easy disposal of waste tires, waste plastics,
2. Experimental details

2.1. Fluff solid refuse fuel (SRF)

Fluff SRF used in this study is a waste plastic solid fuel compressed to a dimension of 50 mm by crushing it into a wrapping paper, bag, plastic, etc. Table II shows the result of the proximate analysis and the ultimate analysis. The results showed that the carbon content had a large proportion (61 %), and H, O, N, S, and Cl had concentrations of 8.59, 17.26, 0.78, 0.41, and 0.79 %, respectively. The LHV was 5,990 kcal/kg.

2.2. Equipment

The system consists of an SRF metering equipment, and a reactor equipped with a thermal plasma torch.

Thermal plasma torch

Figure 1 shows the 3D and actual view of the thermal plasma torch. The plasma torch is heated by the arc gas injected by the arc discharge generated between the cone-shaped cathode tungsten rod and the nozzle-shaped anode, which is designed to be ejected in the form of a high-temperature, high-velocity plasma jet.

SRF feeder

The feeder is a device that can quantitatively input SRF at a rate of 1~5 ton/h without agglomeration; a double screw and a paddle is installed to remove agglomerates.

Gasification and combustion system

The gasification and combustion system is equipped with thermal plasma torches to increase the temperature in the reactor. The SRF is supplied to the combustion reactor via the SRF connection passage in the upper portion, and it has a structure that is discharged by burning and gasifying through the plasma torch region. Inside the reactor, a refractory was installed to prevent damage during high temperatures, and a K-type thermocouple was installed to measure the temperature inside the reactor. For gas analysis, an analysis line was installed at the bottom of the reactor using a sus tube, and the gas analysis was performed using GC (3000 micro GC, Inficon, Switzerland).

| Table II. Characteristics of SRF. |
|-----------------------------------|
| Refuse plastic fuel analysis method | Contents | Valve |
|-----------------------------------|----------|-------|
| Proximate analysis (%)            | Moisture | 7.6   |
|                                   | Volatile | 71.7  |
|                                   | Ash      | 10.8  |
|                                   | Fixed carbon | 9.9 |
| Ultimate analysis (%)             | C        | 60.69 |
|                                   | H        | 8.59  |
|                                   | O        | 17.26 |
|                                   | N        | 0.78  |
|                                   | S        | 0.41  |
|                                   | Cl       | 0.79  |
and gas analyzer (Vario plus, MRU, US). Figure 2 shows the flow chart of thermal plasma systems, while Fig. 3 shows the configuration of the plasma system.

**Experiment method**

SRF is fed into the gasification and combustion system at a feed rate of 2.0 ton/h through the fixed supply system. The process conditions of the thermal plasma torch installed in the reactor are summarized in Table III.

Samples were prepared at the top of the reactor before the temperature in the reactor was raised, and the SRF was added 10 min after plasma discharge. CO, CO2, O2, H2, and CH4 were measured using gas analysis equipment, and NOx, HCl, CO, and CO2 were analyzed by TMS at the final exhaust port.

### 3. Results and discussion

#### 3.1. Simulation

Computational analysis was performed to check the thermal temperature inside the waste plastic treatment system according to the plasma torch output. The computational analysis used ANSYS-FLUENT, a CFD analysis code. The boundary conditions for computational analysis were set up with air and plastic inlets, walls, outlets, and thermal plasma inlets.

Figure 4 compares the temperature distribution in the Z-plane over the entire area at the center of the treatment system, and Fig. 5 shows the X-plane temperature distribution at each point in the treatment system. The results show that as the power increases, the temperature inside the system increases and becomes uniform. When using a high output thermal plasma generator, it is expected that the high temperature region of the thermal plasma will be attracted by the rapid flow in the axial direction, and the temperature inside the system is expected to increase.

#### 3.2. Reactor temperature change

Figure 6 shows the results of measuring the temperature of the combustion reactor four times for 7 h. The temperature sensor is a K-type sensor and it can measure up to 1360 °C. The measurement resulted in a temperature distribution of 1024 – 1360 °C. The temperature inside the reactor, however, is over 1360 °C.

#### 3.3. Pyrolysis gas analysis results

The main reaction to obtain the syngas, which is the main component of CO and H2 from SRF, is largely composed of the partial combustion reaction of the CO and H2 components in the waste with the externally supplied oxygen and the gasification reaction of this combustion product with the unreacted carbon in the SRF.

Table IV lists the exhaust gas analysis table measured at the bottom of the reactor.

| Experiment condition | Power (kW) | N2 gas (lpm) | Feeding rate (ton/h) |
|----------------------|-----------|--------------|---------------------|
|                      | 25        | 100          | 2                   |

**Table III.** Thermal plasma torch working condition.

**Table IV.** Pyrolysis gas generated after pyrolysis.

|          | 1  | 2  | 3  | 4  | 5  | Average |
|----------|----|----|----|----|----|---------|
| CO (%)   | 8.2| 9.3| 10.3| 9.3| 9.2| 9.26    |
| CO2 (%)  | 10 | 9.9| 9.3| 9.9| 9.9| 9.80    |
| O2 (%)   | 0.2| 0  | 0  | 0  | 0  | 0.04    |
| H2 (%)   | 6.2| 7.1| 7.8| 7.1| 7.2| 7.08    |
| CH4 (%)  | 0.2| 0.3| 0.2| 0.2| 0.2| 0.22    |

**Figure 4.** (Color online) Z-plane temperature distribution in the system according to the output change of thermal plasma generator.

**Figure 5.** (Color online) X-plane temperature distribution in the system according to the output change of thermal plasma generator.

**Figure 6.** (Color online) Temperature inside the combustion system.
3.4. Exhaust gas measurement result

The analysis was performed using TMS installed in the chimney. Table V shows the result of NOx, HCl, CO, and CO2 measurements.

4. Conclusions

In this study, temperature and gas composition in the reactor were studied through the gasification and combustion system using fluff SRF. Computational analysis was also performed on the internal environment of the combustion system according to the change in the output of the thermal plasma generator. As the generator output increases, the temperature inside the system increases, and the temperature becomes uniform for the entire area. These results indicate that as the generator’s output power increases, the high-temperature region of the thermal plasma extends to the center of the system as the axial velocity increases. Thus, the high output thermal plasma generator can be expected to attract thermal plasma by the rapid flow in the axial direction and lead to a temperature increase inside the system.

SRF was found to show a high temperature of at least 1024°C above the continuous treatment of 2 ton/h, while CO and H2 concentrations increased based on the gas analysis. The reactor analysis confirmed that combustion and gasification occur simultaneously. In addition, the pollutant gas discharged to the atmospheric gas phase could be treated under environmental regulations. The results confirm that the gasification and combustion composite system can handle various samples.

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