SYNTHESIS, CHARACTERIZATION, AND ANTIBACTERIAL STUDY OF ZINC OXIDE-GRAPHENE NANOCOMPOSITES

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INTRODUCTION

Nanotechnology has emerged as a valuable modus in the pharmaceutical industry as an alternative antimicrobial approach because of arrival of antibiotic-resistant strains of microorganisms [1]. Nanosized particles, either simple or composite by nature, exhibit unique physical and chemical properties and show a potential of being used in various biomedical application [2-9]. There is need to develop uniform nanosize drug particles having precise shape, size, and physical and chemical properties in the production of new pharmaceutical products. The biocidal efficiency of nanoparticles (NPs) may be due to combination of small size and high surface to volume ratio which facilitates intimate interactions with microbial membranes [10,11].

Metal oxides with nanostructure have attracted considerable interest in many areas of technology [12]. Among metal oxide NPs, zinc oxide (ZnO) has received much attention in the recent past. ZnO nanostructures are the forefront of research due to their unique properties and wide applications such as ultraviolet blocking properties [13]. The advantage of using ZnO NPs is that they strongly inhibit the action of pathogenic microbes when used in small concentration [14].

Graphene oxide (GO) which is actually a compact honeycomb structure of sp² hybridized carbon atoms has drawn a huge attention because of its outstanding electronic, thermal, and mechanical properties, which can be applied in nanomedicine field [15-20]. GO can be used to stabilize growing metal NPs and inhibits their aggregation because of its unique properties such as large surface area, low cytotoxicity, and good water stability. On these basis, GO and its composites have a wide range of possible applications on transistors, transparent conductors, polymer reinforcement, bioengineering, and biomaterials areas [21-23].

A lot of research has been carried out on antibacterial properties of Ag NPs and their nanocomposites. In continuation to our earlier study [24-30], in the present study, we have focused on the synthesis of zinc (Zn) NPs by facile, and cost effective sol-gel method and nanocomposite of Zn were prepared with reduced graphene oxide (RGO). Antibacterial properties of the synthesized NPs and nanocomposites were investigated against Gram-positive and Gram-negative bacteria. The aim of this study is to synthesize a nanocomposite material with better or comparable antibacterial performance.

MATERIALS AND METHODS

Materials

All chemicals used were of analytical reagent grade and used without further purification. Chemicals used were graphite powder (<20 µ), ZnCl₂, NaNO₂, H₂O₂, H₂SO₄, KMnO₄, HCl, NaOH, NaBH₄, citric acid, and ethylene glycol. All were available in our research lab. The strains employed in this work were the Gram-negative bacteria (Escherichia coli) and Gram-positive bacteria (Staphylococcus epidermidis). In addition, nutrient broth and agar-agar were used to prepare agar plates.

Preparation of RGO

GO was synthesized from graphite powder by a modified Hummers method [31]. Graphite powder, NaNO₂, and H₂SO₄ were mixed together at 0.0°C. Then, KMnO₄ was added slowly into the reaction mixture with constant stirring. The mixture was heated to 35.0°C and stirred for 12.0 hrs, and then, 500.0 mL of water was slowly added under vigorous stirring. Then, 30.0% H₂O₂ solution was added to reduce the residual MnO₂. The mixture was then washed by acidified water (3.0%) and then with double distilled water three times followed by filtration and drying. RGO sheets were then obtained.
Preparation of ZnO NPs
ZnCl₂ solution and citric acid solution were prepared separately in double-distilled water and were mixed together with continuous stirring for 15.0 minutes. Ethylene glycol was then added into the solution and continuously stirred for 3 hrs. The resultant precipitates thus obtained were washed with double-distilled water and dried at 80.0°C in oven. Finally, these were put into the muffle furnace at 400.0°C for 3.0 hrs. ZnO NPs were thus obtained.

Preparation of ZnO/GO nanocomposites
ZnCl₂ and GO were mixed together in double-distilled water to have a metal oxide loading of 10 wt%. The solution pH was adjusted to 10.0 using NaOH solution and stirred continuously for 4.0 hrs. Then, 50 ml of 0.1 M NaBH₄ was added and stirred continuously for 3.0 hrs. The resulting material was then filtered and washed several times with double-distilled water and dried in oven at 100.0°C in oven for 2.0 hrs. Finally, these were put into the muffle furnace at 600.0°C for 2.0 hrs. ZnO NPs were thus obtained.

Characterizations
The X-ray diffraction pattern of the ZnO NPs and ZnO-GO nanocomposites samples were obtained (Fig. 1) using an X-ray diffractometer (XRD) (Panalytics X-Pert Pro, PU, Chandigarh). The XRD pattern of RGO indicates a broad diffraction peak at 2θ=24°. The broadening and shift of the characteristic diffraction peak of graphite from 26.5° to 24° in RGO which was due to the short-range order in stacked stacks. All XRD diffraction peaks of ZnO powders are shown in good agreement with hexagonal structure of ZnO reported in JCPDS File Card No. 05-0664. Peaks of ZnO at 31.7°, 34.4°, 36.2°, 47.4°, 56.6°, 62.9°, 65.5°, 68.0°, and 69.1° that are corresponding to (100), (002), (101), (102), (110), (103), (200), (112), and (201) lattice planes, respectively, indicating the formation of the wurtzite structure of ZnO NPs. No peaks of impurity are observed, indicating that the high purity ZnO was obtained. The XRD spectra of nanocomposites have peaks corresponding to both RGO and ZnO NPs.

The size, morphology and distribution of ZnO NPs in ZnO-GO nanocomposites were examined using a transmission electron microscopy (TEM) (TECNAL 200 Kv TEM [Fei, Electron Optics], AIIMS, Delhi). Fig. 2 shows TEM images of ZnO NPs, (a) RGO, and (b) ZnO-GO nanocomposite, and the inset of Fig. 2a shows that the ZnO NPs have a spherical shape. The TEM image reveals that the ZnO NPs are dispersed on the GO (Fig. 2c). In addition, the TEM image shows an average particle size of approximately 90 nm for the NPs. From the TEM images, the GO surface looks smooth and integrated (Fig. 2b). In the case of ZnO-GO nanocomposite (Fig. 2c), a large number of ZnO nanocomposites were examined using a transmission electron microscopy (TEM) (TECNAL 200 Kv TEM [Fei, Electron Optics], AIIMS, Delhi).

 RESULTS AND DISCUSSION

Characterizations
The X-ray diffraction pattern of the ZnO NPs and ZnO-GO nanocomposites samples were obtained (Fig. 1) using an X-ray diffractometer (XRD) (Panalytics X-Pert Pro, PU, Chandigarh). The XRD pattern of RGO indicates a broad diffraction peak at 2θ=24°. The broadening and shift of the characteristic diffraction peak of graphite from 26.5° to 24° in RGO which was due to the short-range order in stacked stacks. All XRD diffraction peaks of ZnO powders are shown in good agreement with hexagonal structure of ZnO reported in JCPDS File Card No. 05-0664. Peaks of ZnO at 31.7°, 34.4°, 36.2°, 47.4°, 56.6°, 62.9°, 65.5°, 68.0°, and 69.1° that are corresponding to (100), (002), (101), (102), (110), (103), (200), (112), and (201) lattice planes, respectively, indicating the formation of the wurtzite structure of ZnO NPs. No peaks of impurity are observed, indicating that the high purity ZnO was obtained. The XRD spectra of nanocomposites have peaks corresponding to both RGO and ZnO NPs.

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The size, morphology and distribution of ZnO NPs in ZnO-GO nanocomposites were examined using a transmission electron microscopy (TEM) (TECNAL 200 Kv TEM [Fei, Electron Optics], AIIMS, Delhi). Fig. 2 shows TEM images of ZnO NPs, (a) RGO, and (b) ZnO-GO nanocomposite, and (c) the inset of Fig. 2a shows that the ZnO NPs have a spherical shape. The TEM image reveals that the ZnO NPs are dispersed on the GO (Fig. 2c). In addition, the TEM image shows an average particle size of approximately 90 nm for the NPs. From the TEM images, the GO surface looks smooth and integrated (Fig. 2b). In the case of ZnO-GO nanocomposite (Fig. 2c), a large number of ZnO nanocomposites were examined using a transmission electron microscopy (TEM) (TECNAL 200 Kv TEM [Fei, Electron Optics], AIIMS, Delhi).

Table 1: ZOI (mm) shown by different standard antibiotics with
Staphylococcus epidermidis and Escherichia coli

| Bacterium                  | Standard antibiotics |
|----------------------------|----------------------|
|                            | TE 25 | C 25 | P | AMP 10 | S 10 | S3 300 |
| *Staphylococcus epidermidis* |       |      |  |       |     |       |
|                            | 19 | 17 | 15 | NS | 23 | NS |
| *Escherichia coli*         |       |      |  |       |     |       |
|                            | 11 | 9 | NS | NS | 13 | NS |

ZOI: Zone of inhibition

Fig. 1: X-ray diffractometer of zinc oxide nanoparticles, reduced graphene oxide, and zinc-graphene oxide nanocomposite

Fig. 2: Transmission electron microscopy images of (a) zinc oxide nanoparticles, (b) reduced graphene oxide, (c) zinc-graphene oxide nanocomposite

Fig. 3: Fourier transform infrared spectra of zinc oxide nanoparticles, reduced graphene oxide and zinc oxide graphene oxide nano-composites

Fig. 4: (A) Zone of inhibition (ZOI) produced by different standard antibiotics with *Staphylococcus epidermidis* (B) and (C) different concentration of zinc oxide nanoparticles (ZnO-NPs) and ZnO-graphene oxide (ZnO-GO) nanocomposites. ZOI produced with *S. epidermidis*, (a) ZnSO₄, (b) 100 ppm concentration of ZnO-NPs, (c) 500 ppm concentration of ZnO-NPs, (d) 1000 ppm concentration of ZnO-NPs, (e) ZnSO₄ and graphene, (f) 100 ppm concentration of ZnO-GO nanocomposites, (g) 500 ppm concentration of ZnO-GO nanocomposites, (h) 1000 ppm concentration of ZnO-GO nanocomposites

Fig. 5: Zone of inhibition (ZOI) produced by different standard antibiotics with *Escherichia coli* (A) and different concentration of zinc oxide nanoparticles (ZnO-NPs) (B) and ZnO-graphene oxide (GO) nanocomposites (C) with bacteria. ZOI produced with *E. coli*, (a) ZnSO₄, (b) 100 ppm concentration of ZnO-NPs, (c) 500 ppm concentration of ZnO-NPs, (d) 1000 ppm concentration of ZnO-NPs, (e) ZnSO₄ and graphene, (f) 100 ppm concentration of ZnO-GO nanocomposites, (g) 500 ppm concentration of ZnO-GO nanocomposites, (h) 1000 ppm concentration of ZnO-GO nanocomposites
uniformly on the surface of the GO. The high-magnification TEM image (Fig. 2c) further reveals that ZnO-GO nanocomposites are almost spherical in shape.

The chemical functional groups of ZnO NPs and ZnO-GO nanocomposites were characterized using attenuated total reflectance Fourier transform infrared (FTIR) spectrometer (Perkin Elmer - Spectrum RX-I FTIR, PI. Chandigarh). Fig. 3 shows FTIR spectra of ZnO NPs, RGO, and Zn-GO nanocomposites. In the FTIR spectrum for RGO, the peaks at 1731, 1625, and 1183 cm\(^{-1}\) are assigned to the \(\text{C}=\text{O}\) stretching, \(\text{C}–\text{C}\) stretching, and \(\text{C}–\text{O}\) stretching, respectively. The broad peak at 3250 cm\(^{-1}\) in the FTIR spectrum of ZnO NPs/GO nanocomposite might be attributed to the \(\text{O}–\text{H}\) stretching vibration of absorbed water molecules. The following functional groups were identified; \(\text{O}–\text{H}\) stretching vibrations (3240-3300 cm\(^{-1}\)), \(\text{C}=\text{O}\) stretching vibration (1720-1740 cm\(^{-1}\)), \(\text{C}–\text{C}\) from un-oxidized sp\(^2\)-C=O bonds (1590-1620 cm\(^{-1}\)), and \(\text{C}–\text{O}\) vibrations (1250 cm\(^{-1}\)) in the FTIR spectrum of ZnO-GO nanocomposites which confirms the formation of nanocomposites.

### Antibacterial study

The different researchers have reported antibacterial property of different metal NPs [32-35]. We have tested antibacterial activity of the ZnO NPs and ZnO-GO nanocomposites on Gram-positive (\textit{S. epidermidis}) and Gram-negative (\textit{E. coli}) bacteria using agar well diffusion method. Table 1 shows zone of inhibition (ZOI) (mm) shown by different standard antibiotics with \textit{S. epidermidis} and \textit{E. coli}. Table 2 shows antibacterial effect of ZnO NPs and ZnO-GO nanocomposites against Gram-positive (\textit{S. epidermidis}) and Gram-negative (\textit{E. coli}) bacteria are indicated by measuring the diameter of ZOI (mm). Fig. 4 shows that ZOI produced by different standard antibiotics with \textit{S. epidermidis} (a) and different concentration of ZnO-NPs and ZnO-GO nanocomposites (b) and (c). Fig. 5 shows that ZOI shown by different standard antibiotics with \textit{E. coli} (d) and different concentration of ZnO-NPs (e) and ZnO-GO nanocomposites (f) with bacteria.

Agar plates were prepared using nutrient broth and agar-agar. The wells of 8.0 mm diameter were punched with the help of steel borer into the agar having the test microorganism (at concentration about 5x10\(^{8}\) CFU/ml). The wells were filled with 100.0 µl of ZnO NPs and ZnO-GO nanocomposites of different concentration. A range of standard antibiotics (Hexa disc) was also used as the control. After 24.0 hrs incubation at 37.0°C, the diameters of the inhibition zones were measured against the test microorganisms and optical images were documented by a high definition optical camera.

The highest ZOI shown by standard antibiotics is 23.0 and 13.0 for \textit{S. epidermidis} and \textit{E. coli}, respectively. The highest ZOI shown by ZnO NP and ZnO-GO nanocomposites is 28.0 and 23.0 for \textit{S. epidermidis} and \textit{E. coli}, respectively. Hence, both ZnO and ZnO-GO nanocomposites show better antibacterial properties than six standard investigated antibiotics.

### CONCLUSION

ZnO NPs, RGO, and ZnO-GO nanocomposites have been prepared through facile and easy sol-gel methods. ZnO NPs samples exhibit good antibacterial activities against Gram-negative bacterial strain \textit{E. coli} and Gram-positive strain \textit{S. epidermidis}, but ZnO-GO nanocomposites exhibit better antibacterial activity. The increase of concentration of both ZnO NPs and ZnO-GO nanocomposites results in increase in antibacterial activity. Antibacterial property of both, i.e., ZnO NPs and ZnO-GO nanocomposites was compared with six standard antibiotics. The highest ZOI shown by standard antibiotics is 23.0 and 13.0 for \textit{S. epidermidis} and \textit{E. coli}, respectively. Highest ZOI shown by ZnO NP and ZnO-GO nanocomposites is 28.0 and 23.0 for \textit{S. epidermidis} and \textit{E. coli}, respectively. Hence, both ZnO and ZnO-GO nanocomposites show better antibacterial properties than six standard investigated antibiotics.

### Table 2: Antibacterial effect of ZnO NPs and ZnO-GO nanocomposites against Gram-positive (\textit{Staphylococcus epidermidis}) and Gram-negative (\textit{Escherichia coli}) bacteria is indicated by measuring the diameter of ZOI (mm)

| Bacterial test organism         | Sample concentration in µg/ml | Sample concentration in µg/ml |
|---------------------------------|-------------------------------|-------------------------------|
|                                 | ZnSO\(_4\) | ZnO NPs | ZnSO\(_4\)+graphene | ZnO-GO nanocomposites |
|                                 | 1000 | 100 | 500 | 1000 | 1000 | 100 | 500 | 1000 |
| \textit{Staphylococcus epidermidis} | 0 | 12 | 17 | 19 | 0 | 12 | 17 | 19 |
| \textit{Escherichia coli}        | 0 | 16 | 19 | 21 | 10 | 17 | 20 | 23 |

\textit{ZnO NPs}: Zinc oxide nanoparticles, \textit{ZnO-GO}: Zinc oxide-graphene oxide, \textit{ZOI}: Zone of inhibition.

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