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Growth Diagram and Magnetic Properties of Hexagonal LuFe2O4 Thin Films

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I. INTRODUCTION

Multiferroics have attracted great attention recently because of their promising new functionality and intriguing fundamental science. A multiferroic material with a large ferroic polarization, high ordering temperature, and strong coupling between the ferroic orders is ideal for applications. So far, those desired properties have not been realized in a single phase material. Multiferroics like BiFeO$_3$ where the magnetic and electric orders originate from different parts of the structure have high ordering temperatures but weak coupling between different orders. Other materials, like TbMn$_2$O$_5$, exhibiting ferroelectricity due to the broken symmetry caused by the spiral magnetic moment have strong magnetoelectric coupling. However, here the ordering temperature is very low and the electric polarization is small. LuFe$_2$O$_4$ contains layers of Fe$_2$O$_2$ with a triangular lattice that are sandwiched by LuO$_2$ layers. Combined with the mixed valance of Fe, the Fe$_2$O$_2$ layers in the triangular lattice form a charge ordered state at $T_{CO} = 320$ K, followed by a ferrimagnetic order at $T_N = 240$ K (Ref. 3). Significant changes in dielectric properties have been observed upon application of a small magnetic field at room temperature. The relatively high transition temperature, large polarization, high magnetic coercivity, and the strong magnetoelectric coupling make LuFe$_2$O$_4$ a unique multiferroic material. Recently, the possibility of fast switching and high tunability of LuFe$_2$O$_4$ due to the electronic origin of its charge order was demonstrated.

Compared to the large amount of effort to study bulk LuFe$_2$O$_4$, there are only a couple of reported attempts to grow LuFe$_2$O$_4$ thin films on $\alpha$-Al$_2$O$_3$ (001) and on Si substrates using pulsed laser deposition (PLD). Liu et al. found that the growth of LuFe$_2$O$_4$ on $\alpha$-Al$_2$O$_3$ (001) (with a target consisting of a sintered mixture of Lu$_2$O$_3$ and Fe$_2$O$_3$) needs substrate temperatures as high as 850 $^\circ$C. In addition, a significant deviation of the Lu:Fe stoichiometry from 1:2 was observed, which was attributed to different ablation efficiencies of Lu and Fe in the target. This problem was circumvented by enriching the Fe concentration of the target material. However, as a result, Fe$_3$O$_4$ and Fe$_2$O$_3$ impurities were introduced as intermediate layers between the LuFe$_2$O$_4$ film and the $\alpha$-Al$_2$O$_3$ substrate.

In this paper, we present a comprehensive study on the growth of Lu-Fe-O compound thin films on MgO (111) substrates using pulsed laser deposition (PLD). The experimentally constructed growth diagram shows that the parameter space for growing epitaxial LuFe$_2$O$_4$ thin films turns out to be a narrow window of temperature and O$_2$ pressure conditions. An understanding of the growth mechanism of Lu-Fe-O compound films is offered in terms of the thermochrometry at the surface. Superparamagnetism is observed in the LuFe$_2$O$_4$ film and is explained in terms of the effect of the impurity hexagonal LuFeO$_3$ (h-LuFeO$_3$) phase and structural defects.

I. EXPERIMENTAL CONDITIONS

Lu-Fe-O compound thin films were grown using PLD with a KrF ($\lambda = 248$ nm) laser. The energy density of the laser is 2.5 J cm$^{-2}$ with a repetition rate of 1 Hz. The target-substrate distance is 3.5 cm. The thickness of the films grown in this study is approximately 100 nm. The substrates are MgO (111) single crystals annealed in O$_2$ for 24 hours at 1100 $^\circ$C. The target material used is polycrystalline LuFe$_2$O$_4$, whose
properties are verified using powder x-ray diffraction (XRD) and a superconducting quantum interference device (SQUID). After growth, the sample heating is turned off so that the sample cools to below 200 °C at the same pressure as that of the growth condition within 5 minutes. The substrates were clamped on a heater with a Pt foil in between. The sample temperature was measured by a pyrometer using emissivity of 0.3. In principle, all the growth parameters described above will have to be scanned and optimized to realize the growth of high quality LuFe2O4 thin films. In this work, we are more focused on elucidating the mechanism of the growth. Therefore, fine scans of the substrate temperature and the O2 pressure were carried out to map out the growth diagram involving the growth of more than one hundred samples, while all the other parameters were kept constant.

The growth was monitored using the reflection high energy electron diffraction (RHEED) from the heating up of the substrates before the growth to the cooling down after the growth. The RHEED images of the films in the paper were all taken right after the growth at the growth temperature. Transmission electron microscopy (TEM) work was carried out on a Cs-corrected FEI Titan 80/300-kV TEM/STEM. TEM specimen was prepared with traditional mechanical polishing followed by ion milling. High resolution Z-contrast images were acquired using 300KV and a beam size of 0.7 Å. The magnetic properties of the films were measured using SQUID. The subtraction of background is done by assuming that the magnetization saturates in a large field at 300 K.

### III. RESULTS

#### A. Growth diagram

In this work, we start from the ternary phase diagram of the bulk Lu-Fe-O system, a section of which is shown in Fig. 1(a) at 1200 °C (Refs. 8 and 9). This system belongs to the D-type of lanthanoid-Fe-O compounds for which there are four stable three-element phases: LuFe2O4 (A), Lu2Fe2O7 (B), LuFeO3 (perovskite or P), and Lu1Fe2O12 (garnet or G). In principle, one way to form a single LuFe2O4 phase is to keep atomic ratio Lu:Fe = 1:2 and vary the O2 pressure, shown as a thick dashed line in Fig. 1(a).

To elucidate the mechanism of the growth of Lu-Fe-O compound films, we carried out fine scans of the substrate temperature and the O2 pressure to map out the growth diagram. Figure 1(b) is the resulting experimental growth diagram. The important observations can be summarized as follows. (1) In the low temperature region the growth follows more or less the behavior predicted by the bulk phase diagram Fig. 1(a): at high pressure, the existing phases are LuFeO3, Lu2O3, and h-LuFeO3; when the pressure is decreased, the Fe3O4 phase starts to appear. This is consistent with the fact that LuFe2O4 and Lu2Fe2O7 phases are not stable at low temperature. (2) In the high temperature region, the growth deviates strongly from the thick dashed line in the bulk phase diagram Fig. 1(a) in that the Lu:Fe stoichiometry differs dramatically from that of the polycrystalline LuFe2O4 target. As shown in Fig. 2(a), the LuFe2O4 phase is not observed in the films grown at 1050 °C. The dominant phase is always Lu2O3 (L). The concentration of LuFeO3 rises with increasing O2 pressure. At high enough O2 pressure, h-LuFeO3 compounds start to form. The formation of Lu-Fe-O compounds in the films qualitatively follows the thick dashed line in this narrow window.

#### B. Structural characterization

The combination of in situ structural characterization using RHEED and ex situ characterization by XRD allows assignment of the epitaxial relation between the existing phases and the substrates. The results are given in Table I.
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FIG. 2. (Color online) XRD data of the Lu-Fe-O films. (a) XRD data of the films grown at \(T = 1050^\circ\text{C}\) for two different \(O_2\) pressures. (b) The ratio between the intensity of the P phase (LuFeO\(_3\)) \(I(P)\) and L phase (Lu\(_2\)O\(_3\)) \(I(L)\) as a function of the \(O_2\) pressure at \(T = 1050\) \(^\circ\text{C}\). The line is a fit with the thermochemistry model (see text). (c) Typical XRD data of a LuFe\(_2\)O\(_4\) film that shows the LuFe\(_2\)O\(_4\) \(c\) axis to be perpendicular to the substrate surface as expected. Also present is an h-LuFeO\(_3\) as an impurity phase.

From the RHEED image, one can measure the in-plane lattice constant for the grown film. If three-dimensional (3D) island growth occurs, the RHEED images correspond to the diffraction pattern of the transmitted electron beam which contains more structural information. Figure 3 shows the RHEED images of the MgO (111) substrates and the LuFe\(_2\)O\(_4\) films with the electron beams directed along MgO [1-10] or MgO [11-2]. The strong LuFe\(_2\)O\(_4\) (003), (006), and (009) peaks observed in Fig. 2(b) indicate that the epitaxial relation is LuFe\(_2\)O\(_4\) [001]//MgO [111], [100]//MgO [1-10-2]. The strong LuFe\(_2\)O\(_4\) (003), (006), and (009) peaks observed in Fig. 2(b) indicate that the epitaxial relation is LuFe\(_2\)O\(_4\) [001]//MgO [111], [100]//MgO [1-10]. The line is a fit with the thermochemistry model (see text). The in-plane ratio between the intensity of the P phase (LuFeO\(_3\)) \(I(P)\) and L phase (Lu\(_2\)O\(_3\)) \(I(L)\) as a function of the \(O_2\) pressure at \(T = 1050\) \(^\circ\text{C}\). The line is a fit with the thermochemistry model (see text). The in-plane ratio between the intensity of the P phase (LuFeO\(_3\)) \(I(P)\) and L phase (Lu\(_2\)O\(_3\)) \(I(L)\) as a function of the \(O_2\) pressure at \(T = 1050\) \(^\circ\text{C}\). The line is a fit with the thermochemistry model (see text).

The in-plane lattice constants of the film can be calculated from the separation of the streaks calibrated by the RHEED pattern of the MgO substrate. It is consistent with the LuFe\(_2\)O\(_4\) lattice constant 3.44 Å within the experimental uncertainty of 2%. Hence, the in-plane epitaxial relation is LuFe\(_2\)O\(_4\) [100]//MgO [1-10]. This is unexpected from the point of view of lattice matching, which predicts LuFe\(_2\)O\(_4\) [100]//MgO [11-2] because a \(\sqrt{3} \times \sqrt{3}\) supercell of LuFe\(_2\)O\(_4\) with a 30° rotation along the [001] direction has less than 0.1% mismatch with a 1\(\times\)1 of MgO (111) surface. Contrasting with the apparent quasi-2D growth of LuFe\(_2\)O\(_4\), Lu\(_2\)O\(_3\) forms quasi-3D structures on the substrate. However, the RHEED pattern suggests a face centered cubic structure with a lattice constant half of that of bulk Lu\(_2\)O\(_3\). The detailed structure is not clear at present. At low pressure and low temperature, the RHEED signal is dominated by the diffraction pattern of 3D Fe\(_3\)O\(_4\) islands along the [11-2] direction, with the [111] direction perpendicular to the substrate surface. This is consistent with the XRD data. The lattice constants are the same as that of bulk Fe\(_3\)O\(_4\) within the experimental uncertainty of 2%.

High resolution transmission electron microscopy (HRTEM) reveals the detailed structure of the LuFe\(_2\)O\(_4\) films. As shown in Fig. 4(a), a layered structure of the film is obvious with some variation at different locations. The fast Fourier transforms (FFT) of the HRTEM image at different locations confirm the epitaxial relation observed from RHEED images: the FFT of the substrate [Fig. 4(d)] indicates the reciprocal lattice of MgO viewed from [11-2] direction. The FFT of the majority of the film [Fig. 4(e)] is consistent with the reciprocal lattice of LuFe\(_2\)O\(_4\) viewed from the [1-10] direction, while at some locations [Fig. 4(b)] it suggests h-LuFeO\(_3\) viewed from the [100] direction. These two phases LuFe\(_2\)O\(_4\) and h-LuFeO\(_3\) were further confirmed by direct observation using atomic-resolution Z-contrast imaging, which is shown in Figs. 4(e) and 4(f). The LuO\(_2\)-FeOFeO-LuO\(_2\) ordering in the LuFe\(_2\)O\(_4\) phase and the LuO\(_2\)-FeO-LuO\(_2\) ordering in the h-LuFeO\(_3\) phase are clearly observed.

Although the intensity of the XRD peaks originating from the h-LuFeO\(_3\) phase seems comparable to that of LuFe\(_2\)O\(_4\).
phase, the actual dominant phase is still LuFe$_2$O$_4$ due to the lower x-ray scattering cross section of the LuFe$_2$O$_4$ as compared with that of the h-LuFeO$_3$ phase. This is consistent with the low population of the h-LuFeO$_3$ phase in the HRTEM image. In addition, the RHEED patterns of h-LuFeO$_3$ and LuFe$_2$O$_4$ are supposed to be different according to their structures. The fact that the observed RHEED patterns do not show any indication of h-LuFeO$_3$ within the detection limit also suggests a low concentration of the h-LuFeO$_3$ phase in the films.

C. Magnetic properties

Ferrimagnetism, large magnetization, and giant coercivity are of the key properties of LuFe$_2$O$_4$ (Refs. 19–21). This makes the study of the magnetic properties of LuFe$_2$O$_4$ films critical. As shown in Fig. 5(a), little hysteresis is observed for these LuFe$_2$O$_4$ films, in contrast to the bulk. When magnetization is plotted against magnetic field over temperature ($H/T$), the data of 12, 50, 100, and 150 K fall on top of each other [Fig. 5(b)], indicating superparamagnetic behavior. Assuming superparamagnetism to be present, one can calculate the magnetic moment from the slope of the low field magnetization data with

$$\frac{dM}{dH} = \frac{N\mu^2}{3kT} \mu_0, \quad (1)$$

where $\mu$, $N$, $\mu_0$, and $k$ are the moment of the superparamagnetic domains, number of the domains per unit volume, the vacuum permeability, and the Boltzmann constant, respectively. The magnetic moments normalized to their maximum value as a function of temperature are plotted in Fig. 5(c), which follow the temperature dependence of the bulk saturation magnetization closely, suggesting that the Neel temperature of the films is not very different from the bulk value of 240 K.

IV. DISCUSSION

A. Growth diagram

The most surprising observation of the growth diagram is that the Lu-Fe-O compound formation at high temperature deviates strongly from the Lu:Fe stoichiometry of the target. Here we propose an explanation in terms of competition between nucleation and desorption of adatoms and its dependence on temperature and supersaturation.

The residence time of an adsorbed atom is given by

$$\tau_{ad} = \frac{1}{\nu} \exp \left( \frac{E_{des}}{kT} \right), \quad (2)$$

where $\nu$ is the vibrational frequency and $E_{des}$ is the desorption energy. Clearly, the residence time of an adatom is shorter at high temperature due to the higher desorption rate. The observed loss of Fe atoms suggests a smaller desorption energy (higher desorption rate) for Fe atoms. At low temperature, because $\exp \left( \frac{E_{des}}{kT} \right)$ is large for both Lu and Fe adatoms, the Lu:Fe stoichiometry can be close to that of the target.

The nucleation speed of deposited adatoms is

$$J_{nuc} \propto \left( \frac{\Delta\mu^*}{T} \right)^{1/2} \exp \left( -\frac{\kappa}{\Delta\mu^*kT} \right), \quad (3)$$

where $\Delta\mu^*$ is the effective supersaturation (molar bulk Gibbs free energy change with surface energy consideration), while $\kappa$ is proportional to the square of the edge energy of the nuclei per unit length. Therefore, at the high temperature limit,
the nucleation speed decreases with temperature and a high supersaturation favors a high nucleation speed. Consider the reaction

$$\text{Fe} + \frac{1}{2}\text{Lu}_2\text{O}_3 + \frac{3}{4}\text{O}_2 \rightarrow \text{LuFeO}_3,$$  \hspace{1cm} (4)

which takes place under thermodynamic equilibrium during the annealing time in between the laser pulses, the supersaturation of $\text{O}_2$ is related to the $\text{O}_2$ pressure as

$$\Delta \mu_{\text{O}_2}^{\text{eq}}(T) = \frac{3}{4} N_A k T \ln(P_{\text{O}_2}),$$  \hspace{1cm} (5)

where $N_A$ is the Avogadro constant. Equation (5) suggests that higher $\text{O}_2$ pressure always corresponds to larger supersaturation, resulting in faster nucleation and better Lu:Fe stoichiometry.

Combining Eqs. (5) and (3), one has the analytical relation between the nucleation speed and the $\text{O}_2$ pressure

$$J_{nuc} \propto \left[ \Delta \mu_{\text{O}_2}^{*}(T) + \frac{3}{4} N_A k T \ln(P_{\text{O}_2}) \right]^{1/2} \exp \left[ -\frac{\kappa}{\Delta \mu_{\text{O}_2}^{*}(T) k T + \frac{3}{2} N_A \ln(P_{\text{O}_2})(kT)^2} \right].$$  \hspace{1cm} (6)

Figure 2(b) shows the XRD intensity (peak area) of the LuFeO$_3$ phase relative to Lu$_2$O$_3$ [I(P)/I(L)] as a function of the $\text{O}_2$ pressure at 1050 °C. Assuming that the nucleation speed is proportional to the XRD intensity, one can fit experimental data with Eq. (6). The result shows that $\Delta \mu_0^*(T = 1050^\circ\text{C}) = 269 \text{kJ mol}^{-1}$, similar to the bulk value found as $\Delta \mu_0 = \Delta H^0 - T \Delta S^0 = 258.2 \text{kJ mol}^{-1}$, taking the $\Delta H^0 = -41.8 \text{kJ mol}^{-1}$ and $\Delta S^0 = -121.4 \text{J} \text{mol}^{-1} \text{K}^{-1}$, and $T = 1050^\circ\text{C}$ (Ref. 28).

In the above analysis, the assumptions we made are as follows: (1) the nucleation speed is proportional to the XRD intensity; (2) at high temperature the thermodynamic equilibrium is reached during the annealing between the laser pulses determines the growth. These assumptions appear to be valid because the thermochemical parameters extracted from the model quantitatively agree with those from the literature. In other words, the growth of Lu-Fe-O at 1050°C can be described using equilibrium thermodynamics, presumably due to the thermodynamic equilibration that occurs in between the laser pulses. Here the competition between the desorption and nucleation determines the Lu:Fe stoichiometry. When the temperature is high enough, the time scales of the nucleation and desorption are comparable. In this case, change of nucleation speed (due to the change of supersaturation which is a function of $\text{O}_2$ pressure) affects the Lu:Fe stoichiometry dramatically.

Based on this analysis, we expect the optimal growth conditions for LuFe$_2$O$_4$ films to be a narrow temperature and pressure window considering the necessary high temperature for the stability of LuFe$_2$O$_4$ phase that sets a lower limit, and the different desorption speed of Lu and Fe adatoms which sets an upper limit to the temperature. As we have shown in Sec. III, this is indeed what has been observed in our experiments.

**V. CONCLUSION**

In conclusion, we studied the growth dynamics of LuFe$_2$O$_4$ films on MgO (111) substrates and constructed the growth diagram. According to our understanding, application of the correct thermochemistry is the key to preferential formation of the LuFe$_2$O$_4$ phase: (1) at low temperature, LuFe$_2$O$_4$ is not a thermodynamically stable phase; (2) at high temperature, the Lu:Fe stoichiometry is off by so much due to the faster desorption of Fe adatoms that LuFe$_2$O$_4$ can not be formed; (3) in a narrow range of substrate temperature and $\text{O}_2$ pressure, LuFe$_2$O$_4$ dominates the grown phases with some h-LuFeO$_3$ phase epitaxially sandwiched in between due to the loss of Fe atoms. Superparamagnetism is observed in the film of LuFe$_2$O$_4$ containing h-LuFeO$_3$ impurities. The extracted Neel transition temperature is similar to that of bulk.

This work reveals the growth mechanism of Lu-Fe-O compound thin films, paving the way to the growth of high quality LuFe$_2$O$_4$ thin films and offers an approach to tuning their properties. This will be critical for future applications using LuFe$_2$O$_4$, a unique multiferroic material with large polarizations, high ordering temperatures, and strong magnetoelectric coupling.

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12The peak at 30.27° in Fig. 2(c) is assigned as h(004) based on its proximity to the calculated peak position (Ref. 15) of h(004) (30.52°) and the fact that TEM results are consistent with coexistence of h-LuFeO₃ and LuFe₂O₄. The peak at 29.74° of the upper curve in Fig. 2(a) is assigned as L(222) based on its proximity to the calculated peak position (Ref. 15) of L(222) (29.87°) and the observed peak position of L(222) (29.65°) of the lower curve in Fig. 2(a). However, the dominant contribution should be from L(222) diffraction.

13According to the calculated x-ray powder diffraction intensity of LuFe₂O₄ (Ref. 15), the (0012) peak is two orders of magnitude weaker than the (0015) peak, which explains why the (0015) peak is observed while the (0012) peak is not.

14Several phases do not show up in the XRD data including FeO and Lu₄Fe₂O₁₂. The fact that there are also no RHEED patterns related to these phases suggests that their population is minimal if not zero. The absence of those phases and the presence of the h-LuFeO₃ which is unstable in bulk may be related to the epitaxial preference.

15See Supplemental Material at http://link.aps.org/supplemental/10.1103/PhysRevB.85.155411 for more detailed information of structural parameters, epitaxial relations, analysis of magnetic properties, thermo-chemistry, and representative XRD for each of the growth region in the phase diagram.