Chemical Composition, Cytotoxic and Antimicrobial Activity of Essential Oils from Cassia bakeriana Craib. against Aerobic and Anaerobic Oral Pathogens

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Abstract: The chemical composition of the essential oils from leaves, bark and wood of Cassia bakeriana Craib. was determined by gas chromatography (GC) and gas chromatography–mass spectrometry (GC-MS). Alcohols, aldehydes and fatty acids were the major components in leaf and bark oil, while wood essential oil was rich in fatty acids. Terpenes such as linalool, (E)-nerolidol and phytol were present in low concentrations. The antimicrobial activity against aerobic and anaerobic oral bacteria was evaluated using the microdilution method, as was the cell viability test carried out with Vero cells. The oils from leaves and bark showed high antimicrobial activity, with minimum inhibitory concentrations between 62.5 and 125 µg·mL⁻¹ for most of the tested bacteria, including Streptococcus mutans, the main etiological agent of dental caries. Leaves oil displayed the lowest cytotoxic effect (EC₅₀ of 153 µg·mL⁻¹), while wood oil exhibited the highest toxicity to Vero cells. C. bakeriana oils are thus a source of biologically active compounds against aerobic and anaerobic oral microorganisms. This study is the first report on the chemical composition, antimicrobial activity and cytotoxicity of C. bakeriana.

Keywords: Cassia bakeriana; essential oil; antimicrobial activity; cytotoxicity
1. Introduction

*Cassia bakeriana* Craib. is a tree belonging to the genus *Cassia* and the Fabaceae-Leguminoseae family, native to Thailand. It is also known as pink cassia and can reach 12 to 15 metres in height. Its original climate is humid tropical, but it tolerates well the subtropical conditions with mild winters of Brazil’s south and southeast regions [1]. This plant grows rapidly and shows good adaptation to the Brazilian cerrado. Many species of *Cassia* are used in folk medicine for their different pharmacological properties, such as laxative, purgative, antibacterial, antifungal and antimalarial activities, demonstrating the pharmacological potential of this plant genus. Studies of various parts of these species have reported the isolation of bioactive secondary metabolites, with anthraquinones, flavonoids and other phenolic compounds being the most common constituents present [2]. *C. bakeriana* leaves are considered an alternative source of laxative drugs and they exhibit a significant glycosylated anthraquinone content [3]. The chemical composition of essential oils present in plants has been studied, and these oils are promising sources of secondary metabolites with important biological properties, including antimicrobial activity [4]. A recent review revealed that several essential oils possess strong antimicrobial activity against various microorganisms, suggesting the possibility of using these oils as replacements of synthetic drugs to overcome the increasing resistance of some pathogens [5]. Some works have shown that essential oils are potential antibacterial agents against oral microorganisms [6–9]. These activities are generally related to secondary metabolite substances present in the oils. Several of these compounds were tested separately and have shown significant antimicrobial effects [10]. The chemical composition of the leaves oils of some species of *Cassia* such as *Cassia alata* [11,12] and *Cassia fistula* [13] has been determined.

The microorganisms evaluated in this study are considered to be risk factors for oral diseases such as caries and periodontitis; they can also reach the bloodstream triggering other diseases in the human body, like endocarditis, brain abscesses, throat infections, respiratory and gastrointestinal system infections and bacteremia. The oral cavity is one of the most complex microbiome in the human body. More than 700 bacterial species have been detected in this group by molecular methods [14]. The search for new natural products or prototypes with antibacterial activity would be very important for the control and prevention of oral and systemic diseases in human health. Given this perspective, the aim of the present study was to determine the chemical composition of the essential oils from leaves, bark and wood of *C. bakeriana*, and investigate their antimicrobial potential against aerobic and anaerobic oral pathogens and their cytotoxic effects. This is the first report involving the chemical composition and biological activity of essential oils of *C. bakeriana*.

2. Results and Discussion

2.1. Chemical Composition of the Essential Oils

Table 1 shows the composition of essential oils from the leaves, barks and wood of *C. bakeriana*. Table 2 shows the classification of compounds in the essential oil from different parts of *C. bakeriana* classified by functional groups. The leaves had the highest essential oil yield of 0.94 ± 0.03%, followed by barks at 0.46 ± 0.08%, and wood at 0.12 ± 0.02%. Surprisingly, terpenoids are represented only by linalool (3.55%) and phytol (4.45%) in the leaves’ essential oil.
Table 1. Chemical composition of essential oil from different parts of *C. bakeriana*.

| Compound                        | AI Reference (Calculated) | Identification method | Composition (%) |
|---------------------------------|---------------------------|-----------------------|-----------------|
|                                 |                           |                       | Wood | Bark | Leaves |
| Hexanal                         | 801                       | a, b, c               | 1.13 | 0.82 |        |
| Furfural                        | 828                       | a, b, c               | 1.27 | -    | -      |
| *(E)-Hex-2-enal                 | 846                       | a, b, c               | -    | -    | 10.48  |
| *(Z)-Hex-3-en-1-ol              | 850                       | a, b, c               | 1.56 | 6.73 | 34.90  |
| Hex-2-en-1-ol                   | 854                       | a, b, c               | -    | -    | 0.95   |
| Hexan-1-ol                      | 863                       | a, b, c               | 1.03 | 0.72 | -      |
| Octanal                         | 998                       | a, b, c               | 0.87 | -    | -      |
| Hex-3-en-1-ol, acetate          | 1004                      | a, b, c               | -    | -    | 8.68   |
| Phenylacetaldehyde              | 1036                      | a, b, c               | 3.76 | -    | -      |
| Octan-1-ol                      | 1063                      | a, b, c               | -    | 2.82 | -      |
| Linalool                        | 1097                      | a, b, c               | -    | -    | 3.55   |
| Nonanal                         | 1100                      | a, b, c               | -    | 14.42| -      |
| Nonanol                         | 1165                      | a, b, c               | 0.98 | -    | -      |
| Methyl salicylate               | 1190                      | a, b, c               | 1.27 | -    | -      |
| Octanoic acid (caprylic acid)   | 1192                      | a, c, d               | -    | 1.53 | -      |
| Decanal                         | 1201                      | a, b, c               | 0.89 | -    | -      |
| *Cis*-dec-2-enal                | 1261                      | a, b, c               | -    | 6.17 | -      |
| Nonanoic acid (pelargonic acid) | 1293                      | a, c, d               | -    | 5.77 | -      |
| 4-Vinylguaiacol                 | 1309                      | a, b, c               | -    | 2.41 | -      |
| 4-propylguaiacol                | 1374                      | a, b, c               | 3.92 | -    | -      |
| *(E)-Nerolidol                  | 1561                      | a, b, c               | 4.70 | -    | -      |
| Dodecanoic acid (lauric acid)   | 1565                      | a, c, d               | 1.92 | 0.93 | -      |
| Tridecanoic acid (tridecylic acid) | 1662                  | a, c, d               | 0.79 | -    | -      |
| Tetradecanoic acid (myristic acid) | 1770                  | a, c, d               | 1.45 | -    | -      |
|                                | N. I.                     | -                     | 0.93 | -    | -      |
| Hexadecanoic acid (palmitic acid) | 1959                  | a, c, d               | 58.14| 34.80| 5.89   |
|                                | N. I.                     | -                     | 1.05 | -    | -      |
| Phytol                          | 2114                      | a, c, d               | -    | -    | 4.45   |
| *(Z,Z)-Octadeca-9,12-dienoic acid | 2132                 | a, c, d               | 8.46 | 2.09 | -      |
| (linoleic acid)                 | *(Z)-Octadec-9-enoic acid (oleic acid) | 2132 | 2149 | a, c, d | 15.22 | 2.20 | - |
| Octadecanoic acid (saturated)   | 2158                      | a, c, d               | 3.53 | 0.72 | -      |
| Tricosane                       | 2300                      | a, b, c               | 0.69 | 1.03 | -      |
| Tetracosane                     | 2400                      | a, b, c               | 1.14 | -    | -      |
|                                | N. I.                     | -                     | 2.38 | -    | -      |
| Pentacosane                     | 2500                      | a, b, c               | 1.33 | -    | -      |
| Heptacosane                     | 2700                      | a, b, c               | 8.27 | -    | -      |
|                                | N. I.                     | -                     | 3.62 | -    | -      |
| Octacosane                      | 2800                      | a, b, c               | 0.74 | 2.07 | -      |
| N. I. (hydrocarbon)             | -                         | -                     | 2.06 | 12.57| -      |
| N. I. (hydrocarbon)             | -                         | -                     | 0.93 | -    | -      |

Total identified (%) 98.02 93.39 86.05

AI = Arithmetic index on a DB5 column (comparison with n-alkanes C<sub>8</sub>-C<sub>40</sub>); N. I. = Not Identified. Identification method: (a) Arithmetic index; (b) Adams mass spectral-retention index library; (c) Mass spectrum comparison with the *Wiley Library*; (d) mass spectrum and arithmetic index comparison with NIST Standard Reference Data (*NIST 2011*).
Table 2. Classification of compounds in the essential oil from different parts of *C. bakeriana* by functional groups (N.I. = not identified).

| Functional groups           | Wood     | Bark    | Leaves  |
|-----------------------------|----------|---------|---------|
| Alcohols                    | 2.59     | 11.15   | 37.15   |
| Aldehydes                   | 6.16     | 23.17   | 10.48   |
| Esters                      | -        | -       | 8.68    |
| Oxygenated monoterpenes     | -        | -       | 3.55    |
| Oxygenated sesquiterpenes   | -        | 4.70    | -       |
| Oxygenated diterpenes       | -        | -       | 4.45    |
| Long chain alkanes          | -        | 2.76    | 12.78   |
| Phenolics                   | 3.92     | -       | 2.41    |
| Fatty acids                 | 85.35    | 51.27   | 6.82    |
| N. I. compounds             | 1.98     | 6.61    | 13.95   |

The major components identified in the leaves were (Z)-hex-3-en-1-ol (34.90%), heptacosane (8.27%), hex-3- en-1-ol, acetate (8.6%), (E)-hex-2-enal (10.48%), hexadecanoic acid (5.89%), the oxygenated diterpene phytol (4.45%) and the oxygenated sesquiterpene linalool (3.55%) (Figure 1). Most of these compounds differ significantly from those found in other species of *Cassia*. In previous studies of essential oil of *C. alata* leaves, the major compounds were linalool (23.0%), borneol (8.6%) and pentadecanal (9.3%) [11], and 1,8-cineole (39.8%), β-caryophyllene (19.1%) and caryophyllene oxide (12.7%) [12]. Leaves of *C. fistula* contained as the main constituents phytol (16.1%), tetradecane (10.5%) and hexadecane (8.7%) [13]. The chemical composition of essential oils can vary among species, among the same species and the different plant parts. These differences are due to several factors that affect the oil chemical composition such as climate, soil quality, harvest season and genetics. In general, at low intensity of light the production of monoterpenes decreases and a small daily variation of temperature can stimulate the production of terpenoids [15].

Figure 1. Major components in the leaf oil (A) and bark (B) of *C. bakeriana*.

The wood oil contained fatty acids as the major components, which accounted for 85.35% of the total, including palmitic (58.14%), oleic (15.22%) and linoleic (8.46%) acids as the major ones. Phenylacetalddehyde (3.76%) and 4-propylogyaiacil (3.92%) were also present in significant amounts.

The bark oil was rich in hexadecanoic acid (34.80%), nonanal (14.42%), (Z)-hex-3-en-1-ol (6.73%), (Z)-dec-2-enal (6.17%), nonanoic acid (5.77%) and in the oxygenated sesquiterpene (E)-nerolidol.
(4.70%) (Figure 1). Fatty acids were also the predominant class of compounds in the essential oil of bark at 51.27%.

2.2. Antimicrobial Activity of Essential Oils

The antimicrobial activity of different plant parts against aerobic and anaerobic oral microorganisms was determined, as was their cytotoxicity. The antimicrobial activity and cytotoxicity results are shown in Table 3. The leaves and bark oils indicated strong antibacterial activity, showing inhibition of both types of microorganisms. The MIC values ranged from 62.5 to 125 µg·mL⁻¹ for most of the bacteria tested. All bacteria demonstrated some degree of sensitivity to the essential oils within the concentrations tested. The more susceptible microorganisms were *A. naeslundii*, *P. gingivalis* and *B. fragilis* from anaerobic bacteria and *S. mitis*, *S. mutans*, *S. sanguinis* and *A. actinomycetemcomitans* from aerobic bacteria, while *F. nucleatum* was more resistant. The wood essential oil exhibited lower activity against investigated bacteria with MICs ranging from 500 to 2000 (or higher) µg·mL⁻¹.

### Table 3. Inhibitory effect of essential oils against aerobic and anaerobic oral bacteria and cytotoxic activity of different parts of *C. bakeriana*.

| Microorganisms | Minimum inhibitory concentration (MIC) – µg·mL⁻¹ | Samples of Essential Oils | a CHD |
|----------------|-----------------------------------------------|--------------------------|-------|
|                |                                               | Leaves | Bark | Wood |       |
| Anaerobic      |                                               |        |      |      |       |
| *F. nucleatum* | (ATCC 25586)                                 | 1000   | 1000 | 2000 | 0.922 |
| *A. naeslundii*| (ATCC 19039)                                 | 62.5   | 125  | 1000 | 1.844 |
| *P. gingivalis*| (ATCC 48417)                                 | 125    | 125  | 500  | 3.68  |
| *B. fragilis*  | (ATCC 25285)                                 | 62.5   | 62.5 | 1000 | 1.844 |
| Aerobic        |                                               |        |      |      |       |
| *S. sanguinis* | (ATCC 10556)                                 | 125    | 125  | 1000 | 3.68  |
| *S. mitis*     | (ATCC 49456)                                 | 62.5   | 62.5 | 500  | 3.68  |
| *S. mutans*    | (ATCC 25175)                                 | 62.5   | 62.5 | 2000 | 0.922 |
| *A. actinomycetemcomitans* | (ATCC 43717) | 125    | 125  | 1000 | 14.7  |
| Cytotoxic activity EC₅₀ – µg·mL⁻¹ | Vero cells (ATCC CCL 81) | 153 ± 13 | 119 ± 2 | 93 ± 3 | ---- |

* a CHD: positive control (chlorhexidine dihydrochloride); *b* Gram-positive bacteria; *c* Gram-negative bacteria.

Comparing Tables 1 and 3, the similar results presented by leaf and bark essential oils could be assigned to the identified aliphatic alcohols and aldehydes. On the other hand, fatty acids, which were present in high concentrations in the wood essential oil, apparently had no inhibitory effects on the tested bacteria. Usually, the biological properties of essential oils are determined by the characteristics
of the major components [16]. The antimicrobial activity in this study may be related to a specific metabolite that is in highest concentration in the oil composition or a synergism between major and minor compounds in the mixture. Unidentified compounds in oils can be contributors to the antimicrobial effects by reinforcement or synergistic interactions. Some compounds found in the essential oils are well-known antimicrobial agents, such as \((Z)\)-hex-3-en-1-ol, linalool [17], caprylic acid [18], lauric and linoleic acids [19], aldehydes, particularly \((E)\)-hex-2-enal [20], methyl salicylate [21], \((E)\)-nerolidol [22] and octan-1-ol [23].

Although they did not have a high composition of terpenes, the essential oils of the leaves and bark of \(C.\) bakeriana showed significant antimicrobial activity against the oral pathogens evaluated. Extracts or oils from plant species with MIC values below 100 \(\mu\text{g} \cdot \text{mL}^{-1}\) are considered promising as potential antimicrobial agents [24]. The essential oils of the leaves and barks of \(C.\) bakeriana exhibited strong antibacterial effect against oral microorganisms showing MIC values lower than 100 \(\mu\text{g} \cdot \text{mL}^{-1}\) and lower than other studies in the literature. The leaf oil of \(Lippia\) sidoides and pure standards of monoterpenes (thymol and carvacrol) against \(S.\) mutans, \(S.\) mitis, \(S.\) sanguinis and \(S.\) salivarus, showed MICs from 2,500 \(\mu\text{g} \cdot \text{mL}^{-1}\) to 10,000 \(\mu\text{g} \cdot \text{mL}^{-1}\) [7]. The MIC values of the essential oil of \(C.\) pubescens for the same microorganisms studied in this work ranged from 500 \(\mu\text{g} \cdot \text{mL}^{-1}\) to 2000 \(\mu\text{g} \cdot \text{mL}^{-1}\) [9]. Oils of \(Leptospermum\) scoparium, \(Melaleuca\) alternifolia, \(Eucalyptus\) radiata and \(Lavandula\) officinalis inhibited the growth of \(S.\) mutans, \(S.\) sobrinus, \(A.\) actinomycetemcomitans, \(P.\) gingivalis and \(F.\) nucleatum with MICs ranging from 300 \(\mu\text{g} \cdot \text{mL}^{-1}\) to 10,000 \(\mu\text{g} \cdot \text{mL}^{-1}\) [6].

The essential oil of \(Artemisia\) iwayomogi was evaluated against fifteen oral bacteria, and inhibitions occurred with values between 100 \(\mu\text{g} \cdot \text{mL}^{-1}\) to 3,200 \(\mu\text{g} \cdot \text{mL}^{-1}\), except for \(P.\) gingivalis with an MIC of 50 \(\mu\text{g} \cdot \text{mL}^{-1}\) [8]. Using a concentration of 62.5 \(\mu\text{g} \cdot \text{mL}^{-1}\), leaf and bark oils of \(C.\) bakeriana inhibited the growth of \(S.\) mutans, principal etiological agent of dental caries. The results are even more promising because both oils exhibited higher toxicity to the microorganisms than to Vero cells at the same concentration. A comparison of the cytotoxicity to Vero cells and antimicrobial activity was performed using the selectivity index, which was calculated by the logarithm of the ratio of cytotoxic concentration (EC\(_{50}\)) and the MIC value for microorganisms (SI = log \([\text{EC}_{50}] / [\text{MIC}]\)). A positive value represents higher selectivity against microorganisms than toxicity to Vero cells, and a negative value indicates a higher toxicity to Vero cells than to bacteria [25]. The selectivity index for the leaf oil at inhibitory concentrations of 62.5 \(\mu\text{g} \cdot \text{mL}^{-1}\) and 125 \(\mu\text{g} \cdot \text{mL}^{-1}\) were 0.38 and 0.08 respectively and for the bark oil in the same concentrations were 0.28 and −0.021 respectively. For the oil wood the obtained SI indexes were −0.73 and −1.03 at the concentrations of 500 and 1000 \(\mu\text{g} \cdot \text{mL}^{-1}\) respectively, indicating greater toxicity to Vero cells and low activity against oral pathogens. According to the MIC values and selectivity indexes, the leaf oil showed promising antibacterial activity while simultaneously displaying low cytotoxic effects to Vero cells.

3. Experimental

3.1. Plant Material

Different parts of \(C.\) bakeriana (leaf, bark and wood) were collected in June 2011 at the Federal University of Uberlândia, Minas Gerais, Brazil (18°55′5.95″S; 48°15′34.01″W). The plant is located in
the Cerrado area (savannah). The plant was identified by a specialist, and a voucher specimen was deposited in the Uberlandenses Herbarium, under number 63584, in Uberlândia.

3.2. Essential Oil Isolation

Flesh leaves, barks and wood of *C. bakeriana* were separately cut and about 400 grams of each part were extracted for four hours by hydrodistillation using a Clevenger-type apparatus. The oil obtained was extracted with dichloromethane (5.0 mL), the organic layer was separated and this fraction was dried with anhydrous sodium sulfate, filtered and kept in a closed vial under refrigeration (−10 °C) for further analysis. The percent yield was calculated relative to the dried mass of the initial sample.

3.3. Analysis of the Essential Oils

The oil was analysed by a Shimadzu equipment (Kyoto, Japan) model CG-17A/QP-5000 gas chromatograph coupled to a mass spectrometer (GC-MS), equipped with a 30-m DB-5 capillary column. The carrier gas used was helium at a flow rate of 1 mL/min, detector and injector temperatures were 220 °C and 240 °C, respectively, and the injection volume was 1 µL and the split ratio was 50:1. The oven temperature was programmed from 60 to 240 °C at 3 °C/min. The electron impact energy was set at 70 eV and fragments from 40 to 650 m/z were collected. The quantification of oils were performed in a Shimadzu brand model 2014 GC/flame ionization detector (GC/FID) under the same conditions employed in the GC/MS, using N2/Air/H2 as burning and carrier gas.

3.4. Identification of the Constituents

Identification of each individual constituent of the essential oils was achieved based on Arithmetic Index on DB-5 in reference do standard compounds; and by comparison of their mass spectral fragmentation patterns (NIST database/ChemStation data system and Wiley library/ChemStation data system mass spectral database) [26,27]. For calculation of arithmetic indexes for linear C8-C40 alkanes were calculate by equation based on alkane standards: AI (x) = 100 Pz + 100[(t (x) − t (Pz))/t (Pz+1) − t (Pz)]; where x: compound at time t; Pz: alkane before x; and Pz+1: alkane after x [26].

3.5. Microbial Strains

The tested strains were obtained from the American Type Culture Collection (ATCC, Rockville MD, USA). The following microorganisms were used in the present work: aerobic *Streptococcus mutans* (ATCC 25175), *Streptococcus mitis* (ATCC 49456), *Streptococcus sanguinis* (ATCC 10556) and *Aggregatibacter actinomycetemcomitans* (ATCC 43717) and anaerobic *Fusobacterium nucleatum* (ATCC 25586), *Bacteroides fragilis* (ATCC 25285), *Actonomyces naeslundii* (ATCC 19039) and *Porphyromonas gingivalis* (ATCC 48417).

3.6. Antimicrobial Activity

The MIC (minimal inhibitory concentration) value is the lowest concentration of the compound, extract or oil capable of inhibiting microorganism growth. The essential oils of different parts of *C. bakeriana* were determined in triplicate by using the microdilution broth method in 96-well
microplates [28]. Samples were dissolved in dimethyl sulfoxide (DMSO; Synth, São Paulo, Brazil) at 8,000 µg·mL$^{-1}$, followed by dilution in tryptic soy broth (Difco, Detroit, MI, USA) for aerobic and Schaedler broth (Difco) supplemented with hemin (5.0 µg·mL$^{-1}$) and vitamin K1 (10.0 µg·mL$^{-1}$) for anaerobic; concentrations tested ranged from 2000 to 10 µg·mL$^{-1}$. The final DMSO content was 4% (v/v), and this solution was used as a negative control. The inoculum was adjusted for each organism, to yield a cell concentration of $5 \times 10^5$ colony forming units (CFU) per mL, according to National Committee for Clinical Laboratory Standard (NCCLS) guidelines [29]. The microplates (96 wells) with the aerobic microorganisms were closed with a sterile plate sealer and incubated aerobically at 37 °C for 24 h. The anaerobic microorganisms were closed with a sterile plate sealer and incubated for 48–72 h in an anaerobic chamber (Don Whitley Scientific, Bradford, UK), in 5%–10% H$_2$, 10% CO$_2$, 80%–85% N$_2$ atmosphere at 37 °C. After that, resazurin (30 µL) in aqueous solution (0.01%) was added to the microplates, to indicate microorganism viability for the determination of minimal inhibitory concentration [28]. Chlorhexidine dihydrochloride (CHD) was used as a positive control, and the concentrations ranged from 0.0115 µg·mL$^{-1}$ to 5.9 µg·mL$^{-1}$. The controls of sterility of TSB and Schaedler broths, control culture (inoculum), chlorhexidine dihydrochloride sterility, sterility of the extracts and control DMSO were performed.

3.7. Cytotoxic Activity

Samples of the essential oils were dissolved in methanol and diluted in culture medium DMEM supplemented until form a solution with a concentration of 640 µg·mL$^{-1}$. The cell viability test was performed with Vero ATCC CCL 81 cells (kidney fibroblasts African green monkey). For evaluation of cytotoxicity was used microplate dilution method. A solution containing $1 \times 10^6$ cells in 10 µL supplemented DMEM was prepared, and 100 µL of this solution was pipetted into each well and then the plate was incubated for 6 h at 37 °C with humidified atmosphere and 5% CO$_2$, occurring cell adhesion in the well. Once attached, the culture medium was removed and added solutions of the samples at concentrations of 512, 256, 128, 64, 32, 16, 8 and 4 µg·mL$^{-1}$, starting from the initial solution of DMEM. The final volume in each well was 100 µL and the concentration of cells present in each well was $1.10^4$ cells. The final concentration of methanol in each well did not exceed 3%. Controls growth, negative (100% lysed cells), solvent (methanol) and samples were also prepared. The plates were incubated for 48 h at 37 °C with humidified atmosphere and 5% CO$_2$. After, 10 µL of developing solution of 3 mM resazurin in PBS was added to each well [30] and the plate was incubated again for 24 h under the same conditions. Readings of absorbance at 590 nm was performed in a microplate spectrophotometer. The assays were carried out in triplicate and the results of the absorbances for each concentration tested were calculated according to the growth control. The EC$_{50}$ (concentration at which 50% of the cells are viable) was calculated by a dose-response graph nonlinear regression [31].

3.8. Statistical Analysis

All data on the biological tests were submitted to treatment ANOVA with a significance level of 5%, using the Tukey method in GraphPad Prism 5.
4. Conclusions

The essential oils of *C. bakeriana* leaves and barks have shown promising activity against oral pathogens, including aerobic and anaerobic bacteria. Saturated and unsaturated aliphatic alcohols and aldehydes and terpenes or synergistic interactions between the major and minor constituents within the oils could be responsible for the observed inhibitory effects. The oil leaves indicated higher selectivity against oral pathogens than toxicity to Vero cells in the concentrations that inhibited the growth of most microorganisms evaluated. Our results indicated that some active components are present in oils, so this makes them particularly interesting for future studies and development of novel antimicrobial agents.

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*Sample Availability*: Samples of the essential oils of *Cassia bakeriana* are available from the authors.

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