Refined solution structure of a liganded type 2 wheat nonspecific lipid transfer protein

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Lipid Transfer Protein*  The refined structure of a wheat type 2 nonspecific lipid transfer protein (ns-LTP2) liganded with L-α-palmitoylphosphatidylglycerol has been determined by NMR. The 15N-labeled protein was produced in Pichia pastoris. Physicochemical conditions and ligandation were intensively screened to obtain the best NMR spectra quality. This ns-LTP2 is a 67-residue globular protein with a diameter of about 30 Å. The structure is composed of five helices forming a right superhelix. The protein presents an inner cavity, which has been measured at 341 Å³. All of the helices display hydrophobic side chains oriented toward the cavity. The phospholipid is found in this cavity. Its fatty acid chain is completely inserted in the lipid transfer protein (ns-LTP2) liganded with L-α-palmitoylphosphatidylglycerol glycero-ol moiety being located on a positively charged pocket on the surface of the protein. The superhelix structure of the protein is coiled around the fatty acid chain. The overall structure shows similarities with ns-LTP1. Nevertheless, large three-dimensional structural discrepancies are observed for the I3 and H4 α-helices, the C-terminal region, and the last turn of the H2 helix. The lipid is orthogonal to the orientation observed in ns-LTP1. The volume of the hydrophobic cavity appears to be in the same range as the one of ns-LTP1, despite the fact that ns-LTP2 is shorter by 24 residues.

Plant nonspecific lipid transfer proteins (ns-LTPs)³ were first isolated from spinach leaves and named based on their ability to mediate in vitro the transfer of phospholipids between membranes (1). ns-LTPs are widely distributed and form a superfamily of related proteins subdivided into two families: the type 1 ns-LTPs (ns-LTP1) and the type 2 ns-LTPs (ns-LTP2) (see Refs. 2 and 3 for review). Both families are multi-gemeric, and more than 150 sequences of plant ns-LTPs are listed in data bases. Only a limited number of proteins have been isolated from plant, and in vitro lipid transfer or binding has been demonstrated for an even more limited number of proteins.

The biological functions of ns-LTP1 have not yet been clearly determined, the most favored hypothesis being a role in the transport of cutin monomers (4, 5) or in plant defense mechanisms (6–8) for ns-LTP1. ns-LTP2 gene expression has been reported in the Zinnia elegans cell differentiation process (9, 10), in barley and rice developing seeds (11, 12), under abiotic stress conditions in barley roots (12), or during nodulation in Vigna unguliculate root hairs (13). However, there is no biological evidence of their function in these different contexts. The recent discovery that some ns-LTPs are pan-allergens of plant-derived foods has brought new interest for their study. Most of the ns-LTP allergens identified so far belong to the ns-LTP1 family (14–17), whereas ns-LTP2 has been reported only as a potent allergen of the pollen of Brassica rapa (18).

The three-dimensional structure of four cereal ns-LTP1s has been determined, i.e. wheat (19), barley (20), maize (21, 22), and rice (23, 24). In addition, seven structures of plant ns-LTP1 in complex with ligands have been determined, including those of maize ns-LTP1 with palmitate (21) or palmitoyl-lyso-phosphatidylcholine (22), barley ns-LTP1 with palmitoyl CoA (25) or palmitate (26), and wheat ns-LTP1 with di-myristoyl-phosphatidylglycerol (27), lys-myristoyl-phosphatidylcholine (28), or prostaglandin B2 (29). All of these data showed that ns-LTP1 are compact single domain proteins whose fold is stabilized by four disulfide bonds. They are characterized by a four-α-helix bundle and a C-terminal region with no regular secondary structure. The most interesting feature of ns-LTP1 structure is the tunnel-like hydrophobic cavity that runs through the molecule and appears as a potential site for lipid binding. Although plant ns-LTP1s exhibit very similar global folds, the shape and size of this hydrophobic cavity vary considerably depending on the protein and/or on the ligand. This clearly indicates a high plasticity of the cavity that is able to accommodate a variety of hydrophobic molecules. In contrast, an antifungal protein extracted from onion seeds that showed a structure similar to those of ns-LTP1 except that the internal cavity, obstructed by several aromatic side chains, is unable to transfer lipids (30).

ns-LTP1s have been studied more extensively than ns-LTP2s that are distinct in terms of primary sequence with less than 30% homology (Fig. 1), size (7 kDa versus 9 kDa), and lipid transfer efficiency. In wheat, both ns-LTP1 and ns-LTP2 have been biochemically characterized, and ns-LTP2 exhibits a higher lipid transfer activity than ns-LTP1 (31). They share the same 8-cysteine skeleton, but their disulfide bond assignment has been shown to be different (32). Because the wheat ns-LTP2 exhibits a lipid transfer activity, one can hypothesize the presence of an hydrophobic cavity. However, whether ns-LTP2s have a fold that is similar to or different from that of ns-LTP1

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‡ The abbreviations used are: ns-LTP, nonspecific lipid transfer protein; LPG, L-α-palmitoylphosphatidylglycerol; HSQC, heteronuclear single-quantum spectroscopy; NOE, nuclear Overhauser effect; NOESY, NOE-edited spectroscopy; TOCSY, total correlation spectroscopy; COSY, correlation spectroscopy; MALDI-TOF, matrix-assisted laser desorption/ionization-time of flight.

This paper is available on line at http://www.jbc.org
remained to be elucidated at the initiation of the present work.
This paper presents the refined solution structure of a recombinant
wheat ns-LTP2 liganded with a C16 lyso-phospholipid.
In the final stages of preparation of this manuscript, the structure
of a rice ns-LTP2 was reported (33) (Protein Data Bank entry 1L6H).

EXPERIMENTAL PROCEDURES

Production/Purification—The Pichia pastoris transformant GS115-Tdltp18-tr5.2 (34) expressing a wheat ns-LTP2 was used for the production of
the 15N-labeled protein. Production was carried out in an Applikon fermentor (400 ml of culture) with 99.4% 15N-labeled ammonium sulfate (Euroisotop) as nitrogen source. Labeled (15NH4)2SO4 (0.9% w/v) was added from the very beginning of the biomass production phase and during the induction phase at 0, 24, and 48 h (0.27, 0.27, and 0.1% w/v, respectively). Production of ns-LTP2 was induced by methanol after 22 h of culture and lasted for 77 h. The amount of secreted protein was estimated by densitometer analysis of SDS gels, using purified recombinant ns-LTP2 as a standard. Folding and accumulation of the recombinant
protein was also directly monitored by 15N HSQC NMR (34). A nonlabeled sample was also produced as described earlier (35).

The protein was purified from culture supernatant by a single-step
procedure (36) using expanded bed chromatography (Streamline SP-XL; Amersham Biosciences), and 650 mg of protein were obtained. The protein was dialyzed—containing 0.1% trifluoroacetic acid. A 0.5-
M solution of this solution was mixed with 0.5 H9262 l of water
and HSQC. MALDI-TOF mass spectrometry experiments were con-
ducted by step titration monitored by 15N HSQC. An improvement of the
quality of several parameters was investigated: temperature (300
K), pH (3.45–6.5), ionic strength (0–1 M), stereochemical conditions (300 W), and the spectra were recorded between 285
and 400 nm. Increasing amounts of ligand (0.1–100 equivalents) were added to a solution of ns-LTP2 (30 μM).

NOESY spectra were recorded for 12 h on a Bruker 600 MHz
spectrometer at 300 K on a 2.7 mM ns-LTP2 solution in 60 mM phosphate buffer containing 1 eq of lipid. HSQC spectra were recorded for 30 min on a Bruker 400 MHz on a 1.1 mM ns-LTP2 solution, 10% D2O. The impact of several parameters was investigated: temperature (300–315 K), pH (3.45–5.6), ionic strength (0–100 mM NaCl and 0–60 mM phosphate buffer), and lipid concentration (0–1.5 eq).

Sequence Alignment and Model Prediction—Primary sequence align-
ments were performed with the following software: CLUSTAL W (37) and Psi-Blast (38). Early three-dimensional models of construction and
three-dimensional comparison were performed using a web metaserver
(bioserv.cbs.cnrs.fr; Ref. 39) with the following threading methods:
3DPSSM (40) and TITO (41).

NMR Spectroscopy—All of the NMR samples were prepared by dissolv-
ing the 15N-labeled and unlabeled recombinant wheat ns-LTP2 in H2O (10% D2O) at a concentration of 2.8 mM in presence of 70 mM phosphate buffer at pH 3.5 and 1 mM NaCl, 1.5 eq of 1-palmitoyl-phosphatidyl-
glycerol (1-palmitoyl-2-Hydroxy-sn-glycero-3-[phospho-rac-(1-gly-
cerol)] sodium salt named LPG) was added to all of the protein prepa-
rations. Optimum phospholipid concentration was determined by a step by step titration monitored by 1H HSQC. An improvement of the
spectrum was observed with increasing lipid concentration from 0 to 1.2 eq, with further lipid addition having no effect on the protein spectrum. The NMR spectra were recorded on a Bruker AMX 600 spectrometer, operating at 599.94 MHz, equipped with a triple resonance inverse probe with a field gradient unit on the z axis. All of the data processing
was performed with the version 4 of the Gifsa software (42).

On the 15N-labeled sample, a three-dimensional HSQC-TOCSY
(450-ms mixing time, 80-Hz acquisition) was collected, a three-dimensional
HSQC-NOESY (200-ms mixing time, 64-Hz acquisition 305.2 K), and a
three-dimensional HSQC-NOESY (200-ms mixing time, 82-Hz acqui-
sition, 305.2 K) were performed. Three-dimensional experiments were
processed by linear prediction along the proton and 15N indirect axes
to obtain at least 256 complex points along each axis. Two two-dimen-

FIG. 1. Primary sequence alignment of various ns-LTP1 and ns-LTP2.

The NMR and Fluorescence ns-LTP2/Lipid Screening—Preliminary ex-
periments revealed that NMR spectra could be significantly improved when
ns-LTP2 interacts with a lipid. To find the best conditions for the
study, several lipids and physicochemical conditions were screened by
fluorimetry as well as by NMR. The interaction with L-alpha-palmitoyl-
phosphatidylglycerol (LPG), L-alpha-palmitoylphosphatidylcholine, dimyr-
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terol was tested by fluorimetry and NMR. Lipids from Avanti poling were
prepared in solution in ethanol at a concentration of 10 mM or in small unilamellar vesicle obtained by sonication in H2O.

The fluorescence experiments were performed on an ISS instrument
equipped with Hamatsu detector at 294 K in a 150-μl volume, excitation
was set at 275 nm (300 W), and the spectra were recorded between 285
and 400 nm. Increasing amounts of lipid (0.1–100 equivalent) were
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to obtain at least 256 complex points along each axis. Two two-dimen-
signal NOESY (200-ms mixing time, 18.5-h acquisition, 295.2 and 305.2 K) were also performed on this sample.

On the unlabeled sample, COSY, NOESY (200-ms mixing time), and TOCSY experiments were performed at the temperatures of 295.2, 310, and 323 K. Additional experiments were also performed in pure D$_2$O. Three NOESY experiments ($\tau_e = 50, 100$, and 200 ms) and a COSY experiment were performed on the unlabeled protein liganded with a perdeuterated dodecylphosphocholine lipid (DPC) as ligand and confirmed the assignment. Deuterium exchange experiments were performed by following the evolution of the TOCSY spectrum on a 3.3 mm sample in D$_2$O (60-ms mixing time, pH 3.3, 1 eq of LPG). To follow the most labile amide protons, the lyophilized protein was dissolved in D$_2$O at a temperature of 273 K. The exchange was then monitored by several experiments, increasing the temperature from 0.5 to 47°C, 1-h acquisition, one every 2 h.

The initial spectrum was acquired at 273 K, starting directly after mixture, and a second one was acquired at the same temperature starting 3.5 h after the mixture. Then several spectra were acquired while raising the temperature to follow less labile protons: at 295.2 K and then initial mixture and at 310 and 321 K after mixture. The sample was then left at room temperature for 4 days, and a final experiment was acquired at 295.2 K. A natural abundance $^{13}$C HSQC experiment was performed in D$_2$O to confirm methylene assignment. Rotating frame NOE-edited spectroscopy experiments were performed at 295.2 K (50 and 100 ms) on the unlabeled protein liganded with LPG and with DPC. Diffusion ordered spectroscopy (43) was performed at 295.2 K (diffusion duration, 200 ms; gradient duration, 1.5 ms, with varying diffusion intensities from 0.5 to 47 G/cm, 1-h acquisition). Diffusion ordered spectroscopy experiments were performed on the unlabeled sample at 290 K, and the stimulated echo-longitudinal eddy current delay sequence (44, 45) was used with a WATERGATE filter applied for water suppression. A set of small molecular globular proteins was measured under the same conditions to determine the molecular mass calibration.2 Several $^{13}$P NMR experiments were performed at 295.2 K on a AMX 400 Bruker spectrometer on a 2 mm sample (pH 3.5, 80 mM NaH$_2$PO$_4$ with 2.5 mM of LPG). Two $^{31}$P exchange spectroscopy two-dimensional experiments (mixing time of 0.2 and 2 s, respectively) did not present any evidence of a bound/free equilibrium. All of the NMR data sets have been deposited in the NMRdb data bank (nmrdb.cbs.cnrs.fr; NMRb number lgtp2a).

NMR Spectra Assignment—The assignment of the wheat ns-LTP2 complexed with the phospholipid was performed from the set of three-dimensional (HSQC-NOESY and HSQC-TOCSY) and two-dimensional experiments (COSY, TOCSY, and NOESY) using the sequential assignment strategy (47) with the help of the Rescue software for the amino acid typing step (48). The assignment module of the Gifa program was used for this purpose (49).

In the HSQC experiment, all of the amide peaks could be found. All of the side chain labile hydrogens from Lys, Asn, Arg, and Gln residues were also assigned. Additional peaks belonging to the glycosylated proteins were sorted out in the HSQC spectrum from their reduced intensity. On the basis of sequential dNN(i+1) and doN(i+1) all of the amino acids have been found, and 98% of all the nonlabile $^1$H chemical shifts have been assigned. The natural abundance $^{13}$C HSQC spectrum was used to help the assignment of side chain methylenes. Stereospecific assignment of $\beta$ protons of methylene groups was performed by examination of the NOE intensity and coupling constant patterns.

Under the experimental conditions and because of the blurring of the NOE spectra caused by the exchange cross-peaks, the phospholipid signal could not be unambiguously assigned, except for the C16-terminal methyl protons. The complete assignment of the $^1$H and $^{13}$N chemical shifts of the wheat ns-LTP2 complexed with LPG has been deposited in the BioMagResBank (number 4977).

NMR Structure Calculation and Analysis—The two-dimensional and three-dimensional NOESY experiments, performed at three different temperatures (295.2, 310, and 323 K), have been used to extract all of the distance constraints used for the structure reconstruction. Peak intensities were corrected against a set of reference peaks, using the standard tools provided with the Gifa program (49). Intensity levels were analyzed with a 1/\textit{n} law, and distance constraints were obtained from the intensities by classifying in long, medium, and short range distances; pseudo-atom corrections were used. Additional constraints were obtained on the $\varphi$ angle by measuring $^3$J NH-Ha couplings constants with nitrogen method (50) against the $\chi_1$ angle. The observed NOESY peak intensity between HN-Ha spins and H$\beta$, H$\beta$, spins, when a stereospecific assignment was available.

The dynamical annealing protocol (anneal.im) of the crystallography NMR (CNS) software (51) was used to generate the protein structure from the set of constraints. A first set of structures was obtained from the constraint list, without any constraint on the cysteine linkage and with no phospholipid in the topology. From this set of structures, the diulfide linkage could be unambiguously assigned, because all of the sulfur atoms were located in compatible distances. The observed disulfide bridges are: Cys-25/Cys-60, Cys-10/Cys-24, Cys-2/Cys-34, and Cys-36/Cys-67. H$\alpha$-H$\alpha$ NOE contacts were further observed in the NOEY map, which confirmed this assignment.

The model of the liganded protein was constructed from the same set of constraints. Additional constraints were used, corresponding to the observed connectivity of the C16 terminal CH3 of the fatty acid chain and the His of Tyr-44 amino terminal chain (C8) with H$\delta$ of Phe-35 with a long distance constraint. No additional constraint was used to force the exit of the phospholipid from the protein core. The distance constraints corresponding to the S-S bounds were then added to the constraint list. The dynamical annealing protocol was used for the calculation of 250 structures, from which the 10 structures with the lowest global energy were conserved. The obtained set of structures was deposited to the Protein Data Bank (code 1N89).

The Procheck (52) program was used to check the quality of the obtained structures, as well as to compute the Ramachandran maps. The VOIDOO (53–55) was used to compute the volume of the inner cavity.

RESULTS AND DISCUSSION

Production and Purification—The recombinant protein preparations were obtained as previously described (34–36). The N-terminal sequence (ACQASQALVC) of the recombinant ns-LTP2, determined by mass spectroscopy, is identical to those of the wheat-purified ns-LTP2 (56), indicating that the recombinant protein was correctly processed by the P. pastoris KEX2 protease.

The mass spectroscopy was performed on the $^{15}$N-labeled protein. A 7055-Da average molecular mass was measured, confirming a $^{15}$N isotopic labeling over 95%. The mass spectrum also revealed that approximately 15% of the protein has been glycosylated with one to four C6 sugar moieties, assumed to be mannose (162 Da). This is consistent with the fact that P. pastoris adds O-glycosylated and glycosylated residues. Peptide mass fingerprinting of the digested protein revealed that only the first tryptic fragment at the N terminus of the protein (fragment 1–29) is glycosylated.

N-Glycosylation requires the Asn-Xaa-Ser/Thr consensus sequence, whereas O-glycosylation requires the presence of a Ser or Thr residue. The wheat ns-LTP2 does not contain the Asn-Ser/Thr sequence in its first 29 residues; however, it does contain four serine residues (at positions 5, 12, 16, and 21) and one threonine residue (at position 27), residues that are accessible for possible glycosylation. The presence of a proline residue in the vicinity of a serine or threonine residue can enhance O-mannosylation (57). Of the four serine residues available for O-linkage, residue 21 is the only one close to a proline residue. All of these results indicated that a small fraction of the wheat ns-LTP2 is expressed as a P. pastoris KEX2 protease.

For the purification, the recombinant protein prep-

Physicochemical Context Screening and Ligand Choice—Intensive screening of the physicochemical solution conditions was performed to find a set of conditions that provides NMR spectra of good quality. Ligand nature and ionic strength conditions were found to be critical.

Fluorescence screening experiments were performed on a series of phospholipids and fatty acids, highlighting a higher affinity for negatively charged phospholipids. We finally selected LPG with an average chain length (C16). The titration of the ns-LTP2 by LPG, monitored by NMR, shows that the ns-LTP2 becomes more structurally constrained in the presence of this ligand. A narrowing and a spreading out of the peaks are observed on the two-dimensional HSQC spectrum. This evolu-

$^2$ P.-O. Schmidt, S. Augé, and M.-A. Delsuc, personal communication.
Fig. 2. HSQC spectrum of 15N-labeled recombinant wheat ns-LTP2. The spectrum has been recorded at 295.2 K on a Bruker AMX 600 on a 2.8 mM sample complexed with 4.2 mM of L-palmitoylphosphatidylglycerol in 70 mM phosphate buffer at pH 3.5. The assignment is indicated by numbering the peaks with the identity of each residue. Unassigned peaks are indicated with question marks.

tion is stabilized around 1 eq of LPG. The addition of 70 mM phosphate buffer was observed to also improve spectral quality (Fig. 2). On the other hand, we found that the protein could adapt a large range of pH (3.45–6.5) and temperatures (273 K to 323 K) without important modification of the HSQC spectra.

Under these conditions, a diffusion ordered spectroscopy experiment was used to determine the translational diffusion coefficient of the protein in the experimental conditions, found at 117 μm2/s. Based on a calibration performed on a series of small globular proteins measured in the same conditions, this value confirms that the sample is monomeric. This result is in good agreement with the relaxation study, which has determined a rotational correlation time compatible with a monomeric form (58).

Under these optimized conditions, the 1H-15N HSQC spectrum of the 15N-labeled ns-LTP2 protein liganded with LPG displays more peaks than expected, considering the number of residues. A second step of purification designed to remove the glycosylated proteins was performed on the sample, and a 1H-15N HSQC was recorded. This spectrum displays a reduced number of peaks, but the peaks are still too numerous (labeled with question marks in Fig. 2).

One-dimensional and exchange spectroscopy two-dimensional 31P spectra were recorded on this sample. Two phosphorus lines can be observed in the one-dimensional spectrum, corresponding to the bound phospholipid and to the slight excess of free phospholipid in solution. No additional peaks nor any exchange peak could be observed in the exchange spectroscopy spectra, even at very long mixing time. To observe eventual chemical exchange, a two-dimensional 1H rotating frame NOE-edited spectroscopy experiment was performed on the LPG-liganded protein. In this spectrum a strong exchange peak is visible in the Hα region, as well as several weaker peaks close to the diagonal in the aliphatic and amide regions. The same peaks are observed when the LPG ligand is replaced with a fully deuterated DPC phospholipid. An exchange between the holo and the apo states of the protein being excluded by the 31P experiments, this is an indication that a conformational equilibrium between a major and a minor form of the protein is taking place. This equilibrium is the source of the additional peaks observed in the HSQC spectra. It was taken into account during the assignment phase, and several peaks were assigned to the minor form. However, no effort was done to fully assign this minor form.

Assignments and Secondary Structure Elements—The assignment of the protein resonances was performed from the set of 1H-15N-edited NOESY and TOCSY spectra. The sequential strategy was used, aided by the 1H and natural abundance 13C HSQC spectra (Fig. 3).

The solvent protection experiments indicate that the secondary structure is quite strongly established, because about 40% of the amide protons remain unexchanged during the first hours of the D2O exchange experiment at 1 °C, and eight remain unexchanged after 5 days at room temperature. The solvent protection patterns, as well as J-coupling and NOESY patterns, are indicative of a structure mostly helical. On the other hand, the secondary structure prediction program Jpred (59) anticipates two helical zones, ranging from residues 22 to 29 and from residues 33 to 39.

Chemical assays as well as mass spectroscopy have shown that all eight cysteines of the protein are engaged in disulfide bridges. We have not been able to unambiguously assign the Hβ-Hβ NOE contacts characteristic of this structure because of the crowding of this spectral region. However, after the first run of structure generation, all of the obtained structures exhibited side chain proximities permitting the disulfide skeleton based on these prestructures to be assigned. The disulfide bridges thus found are: Cys-25/Cys-60, Cys-10/Cys-24, Cys-2/Cys-34, and Cys-36/Cys-67. This is in agreement with the chemically determined assignment (32).

The 1H assignments of the LPG in the complexed form were obtained from two-dimensional COSY homonuclear experiments on the free LPG and by comparing the homonuclear spectra obtained from the LPG-ns-LTP2 complex and from the DPC-ns-LTP2 complex. A few chemical shifts of the fatty acid chain were clearly identified: the terminal methyl group (C16), its vicinal methylene (C15), as well as the proximal methylene (C2); the glycerol moiety attached to the fatty acid chains was also assigned. Some intermolecular NOE contacts were observed.

Structure Determination—From the complete set of geometric constraints extracted from the NMR spectra, a set of 10 structures has been obtained. They do not display any important constraint violation, and all of the residues are localized in the allowed regions of the Ramachandran plot. The ensemble of the 10 best structures present a root mean square deviation computed on residues 2–67 of 0.9 Å for all of the heavy atoms and a root mean square deviation of 0.67 Å for the backbone atoms. All of the statistics of the geometrical constraints and the structure reconstruction are given in Table I. The set of solution structures of ns-LTP2 liganded with LPG as obtained from this experimental work is presented in Fig. 4.

The protein is observed as a globular protein with a diameter of about 30 Å. The structure is composed of five helices arranged in a superhelix tertiary structure. Helix 1 is a 3–10 helix, encountering residues 7–15. All of the other helices are α-helices. Helix 2 includes residues 22–31, helix 3 includes residues 34–40, helix 4 includes residues 44–49, and helix 5 includes residues 51–60. The overall fold is a right superhelix. The localization of the helices is in good agreement with the proton exchange experiment, which has shown that there are five main zones in which the amide protons are protected against solvent exchange: residues 14 and 15, residues 26–31,
residues 38–41, residues 47–50, and residues 56–64. The LPG molecule is found partly embedded in the structure of the protein, with the superhelix structure of the protein coiled around the fatty acid chain, and with the phosphate group and the external glycerol moiety unstructured and located outside of the core of the protein.

Helices 1 and 2 are organized in a near anti-parallel conformation. The contact between helices 1 and 2 is tightened by the Cys-10/Cys-24 disulfide bridge. They are linked by an extended strand from Ser-16 to Gly-22. Lys-19 appears to be oriented toward the solvent and does not display any NOE contact with other residue. It should be noted that in a previous dynamic study, it has been observed that the H-N vector of Lys-19 is highly mobile in the ns range (58). Helices 3–5 form a square configuration. Helix 3 contains a characteristic Cys-34, Phe-35, Cys-36 pattern, with Phe-35 buried into the structure and contributing to the hydrophobic core of the structure. Cys-34 and Cys-36 are respectively engaged in disulfide bridges with Cys-2 and Cys-67, forming two diametrically opposed bonds relative to the helix axis. All of the prolines are observed in trans conformation, as confirmed by the Hα–Hα–Pro(i + 1) contacts observed in the NOESY spectra, for all of the proline residues (Pro-20, Pro-42, Pro-51, and Pro-65).

Several additional secondary structure elements can be observed in the structure. The C-terminus of helix 1 presents an unusual hydrogen bond pattern, with the carboxyl moiety of Ser-12 being engaged with the HN of Ala-18. This structure is found in the 10 structures generated, and the HN of residue Ala-18 is found to be protected against solvent exchange, thus confirming this result. A type-1 γ turn is observed between helices 2 and 3, corresponding to residues Gln-31, Gln-32, and Gly-33. A type-1 β turn is observed between helices 3 and 4, corresponding to residues Asp-41, Pro-42, Thr-43, and Tyr-44, with the acidic head of Asp-41 engaged in a hydrogen bond with the HN of Thr-43. Three classic helix cappings can also be observed: the N-capping of helix 2 with the OH of Ser-21 hydrogen-bonded to the HN of Glu-23; the C-capping of helix 2 with the side chain of Gln-32 bonded to the CO of Arg-29; and the N-capping of helix 5, with the OH of Ser-50 bonded to the HN of His-52. Finally, a transient slat bridge between the amide moiety of Lys-40 and the C-terminus of the backbone can be observed in several structures of the NMR ensemble.

The protein presents an inner cavity, which has been measured at 341 Å³. All of the helices present hydrophobic side chains directed toward the cavity. The phospholipid is found in this cavity. Only one unique phospholipid position is observed in the cavity for all 10 retained structures. The fatty acid chain is completely embedded in the protein structure. Its axis is aligned with the axis of the tertiary superhelix and is orthogonal to the α-helix axes (Fig. 4c). The terminal methyl group is positioned between the H1 and H4 helices. The fatty acid chain is inserted in the cavity constituted by the hydrophobic residues (Leu-7, Leu-28, Phe-35, Tyr-38, Tyr-44, Ile-48, Ala-53, Leu-57, Val-64, and Pro-65). The chain presents a turn near carbon 8 and exits the cavity in a cleft between helices H4 and H5 and the C-terminal residue. The inner glycerol moiety

FIG. 3. Synthetic plot showing the elements used for the structure determination.

TABLE I

| NMR restraints and structural statistics of the ensemble calculated for LPG-liganded wheat ns-LTP2 (10 structures) |
|--------------------------------------------------|
| Distance & angles restraints                      |
| NOE intraresidual                                  | 403 |
| NOE sequential (i, i + 1)                          | 188 |
| NOE medium range (|j| – j| ≤ 3)                     | 72  |
| NOE long range (|j| – j| > 3)                     | 95  |
| NOE Intermolecular                                 | 5   |
| S-S bridges                                       | 4   |
| Dihedral angle φ                                  | 29  |
| Dihedral angle χ₁                                  | 8   |
| Total                                             | 800 |
| Deviation from idealized geometry                 |
| Bonds (Å)                                         | 0.0032 ± 0.0013 |
| Angles                                            | 0.37 ± 0.012 |
| Improper angles                                   | 0.18 ± 0.017 |
| Dihedral angles                                   | 30.16 ± 3.04 |
| Deviation from restraints                         |
| NOE restraints                                     | 0.029 ± 0.001 |
| Dihedral restraints                               | 0.45 ± 0.1 |
| Ramachandran plot (see Ref. 52)                   |
| Most favored regions                              | 74.2% |
| Allowed regions                                    | 23.5% |
| Generously allowed regions                        | 2.4% |
| Disallowed regions                                | 0.0% |
| Mean energies (kcal/mol)                          |
| E_bond                                          | 10.29 ± 0.61 |
| E_angle                                          | 37.14 ± 2.5  |
| E_dihedral                                      | 2.67 ± 0.74  |
| E_dihedral_capping                                | 23.5 ± 3.8   |
| E_angle_capping                                  | 48.5 ± 3.93  |
| E_angle_capping                                   | 0.94 ± 0.39  |
| Root mean square deviation from average structure (Å) (residues 2–67) |
| Backbone                                          | 0.67 |
| Heavy atom                                        | 0.90 |
is located on a pocket on the surface of the protein. This pocket presents a basic environment constituted by the Arg-49, Arg-54, and His-66 side chains. These basic residues are observed in close proximity with the phosphate moiety, equilibrating the phosphate charge (Fig. 5). The cavity of ns-LTP2 is asymmetric. The proximal entrance of the cavity, where the phosphate group is found, presents several hydrophilic and basic groups: Arg-49, Arg-54, Thr-58, and His 66. The distal opening of the cavity is characterized with hydrophobic residues, such as Leu-7, Tyr-38, Tyr-44, and Tyr-47.

This location is in good agreement with the following spectroscopic observations: In the fatty acid chain only three unambiguous NOE intermolecular contacts have been identified. Four NOE contacts connect the LPG-terminal methyl groups (C16) with the Tyr-44 and Tyr-47 aromatic part. The other NOE connects the one of the methylenes of the fatty acid chain with the Hβ of Phe-35. All of these residues are found in the hydrophobic cavity. No NOE contacts were found between the protein and LPG glycerol moiety. The chemical shifts of the phospholipid glycerol moiety do not present much shift upon complexation with the protein. On the other hand, the fatty acyl chain chemical shift differences between isolated and liganded LPG are important.

The structure of the minor form present in solution has not been studied, even though it appears to be structured. About one-third of the amino acids seem to be involved in the conformation equilibrium; no attempt was made to assign the residues involved in this equilibrium. Previous dynamic study (58) has shown that the major form is predominantly rigid, with $S^2$ ranging from 0.8 to 0.9, except for Lys-19, exposed to the solvent.

**ns-LTP2 versus ns-LTP1 Comparison**—The three-dimensional structure is known for several ns-LTP1 from different species (19–29). They are very similar among plant and consist in four α-helices organized in a superhelix structure and connected by four disulfide bridges. We compared the structure we present here with wheat ns-LTP1 (Protein Data Bank code 1gh1) (19) using the Visual Molecular Dynamics software (46) (Fig. 6).

The best superposition is obtained when the H1, H2, and H5 (residues 3–16, 22–32, and 50–63) helices in ns-LTP2 are superimposed with the H1, H2, and H4 helices (residues 5–8, 10–
The H2 helix is also different in both structures. Blue charged residues are colored seen how the terminal glycerol group sticks out of the molecular surface and how the basic residues surround the phosphate group (in yellow). The orientation is similar to that in Fig. 4a.

19, 23–35, and 63–76 in ns-LTP1. This gives a root mean square deviation of 1.88 Å (backbone only). The global structure is similar and consists in an hydrophobic cavity structure adapted to the lipid transport. Nevertheless, large three-dimensional structural differences between ns-LTP1 and ns-LTP2 are observed in two regions: the H3 and H4 (respectively H3) α-helices and the C-terminal region. The last turn of the H2 helix is also different in both structures.

The two short H3 and H4 α-helices orthogonal to each other in ns-LTP2 are replaced by the longer H3 helix in ns-LTP1. ns-LTP1 H3 helix is oriented differently and flanked by two large loops presenting no regular secondary structure. This structural difference opens up the cleft corresponding to the entrance of the cavity in which the ligand is located. The C terminus of ns-LTP1 presents a large loop with no secondary structure. The corresponding zone is deleted in ns-LTP2, thus releasing access to the hydrophobic cavity. This deletion has globally no impact on the position of the disulfide bridge involving the last cysteine (Cys-67), which is conserved. As a consequence of these structural changes, the wheat ns-LTP2 is smaller and more globular than ns-LTP1.

All of the cysteine residues are involved in disulfide bridges, but the wheat ns-LTP2 differs from ns-LTP1 in the way the connections are made between the cysteines. ns-LTP1s exhibit a conserved CXC motif located in helix 3, with X being an hydrophilic residue. The two cysteines flanking this hydrophilic residue are linked to cysteines at distal positions in a crossed scheme (as showed in Fig. 1). In the wheat ns-LTP2, an hydrophobic residue (Phe-35) is present, and the two flanking cysteines are involved in disulfide bridges in a noncrossed scheme. The hydrophobic side chain of Phe-35 is displayed on the cavity surface of the wheat ns-LTP2, whereas in ns-LTP1 the corresponding hydrophilic residue is exposed to the solvent, because of a 180° rotation of the entire helix 3 along its main axis.

This cysteine linkage pattern is not unique to ns-LTP2 and is observed in the soy bean hydrophobic protein. Searching structural data bases (40) for a protein presenting a similar structural arrangement to ns-LTP2, the soy bean hydrophobic protein (Protein Data Bank code 1hyp) is found as the best match. This protein shares some structural similarity with ns-LTPs but does not transfer lipids. It presents an analogous organization in terms of helix number and orientation, as well as disulfide bridge topology. Helix 3 is involved in a noncrossed disulfide bridge scheme and appears not to be amphiphilic when compared with ns-LTP2. This could explain why soy bean hydrophobic protein does not exhibit lipid transfer capability.

The presence of an hydrophobic cavity is a characteristic of ns-LTPs; the hydrophobic ligands are bound in this cavity in a rather nonspecific manner. No major differences have been observed between the structure of the free and palmitate complexed maize (21) and barley (25) ns-LTP1 and free and prostaglandin B2-ligated wheat ns-LTP1 (29), whereas large conformational changes have been seen for barley ns-LTP1 when it complexes with palmitoyl CoA (25). Orientation of the lipid within the hydrophobic cavity was found to be opposite in maize (21) and barley (26) liganded structures, whereas wheat ns-LTP1 is able to bind two monoacylated lipids insert head to tail in the hydrophobic cavity (28).

Measured volumes of this hydrophobic cavity are highly variable in ns-LTP1 (23), with typical values ranging from 150 to 580 Å3. However, barley ns-LTP1 exhibits a large volume change upon palmitoyl-CoA binding, with a measured cavity volume increasing from 39 to 620 Å3. The volume observed for the wheat ns-LTP2 appears to be in the same range, even though the protein is smaller by 24 amino acids.

The orientation of the phospholipid main chain, observed in ns-LTP2, is roughly orthogonal to the α-helix axes, and the chain runs from helices H1 to H5. This is in contrast with most described ns-LTP1 cavities found with a main axis parallel to the α-helix axes.

Conclusion—The present work presents the refined structure of the wheat ns-LTP2 protein, liganded with 1-α-palmitoylphosphatidylglycerol, as determined by NMR spectroscopy. The protein was observed as being composed of five helices, structured as a right superhelix, surrounding the phospholipid. This structure presents some homologies with other lipid transfer proteins such as ns-LTP1; however, the phospholipid was found in a quite different location than in most LTP1s. An hydrophobic cavity was also observed, with a volume equivalent to the one found in ns-LTP1 but with a different geometry.

The ns-LTP2 protein structure also presents homology with the one soy bean hydrophobic protein that does not exhibit any lipid transfer activity. This permits us to devise a protein family encompassing ns-LTP1 and ns-LTP2 but also soy bean hydrophobic protein and other related vegetal proteins, based on structural homologies, rather than on the function of primary sequence homologies.

Extensive ligand screening had to be undertaken to find a set of conditions allowing structural study. This indicates that even if the ns-LTP2 is able to adapt a large number of hydrophobic ligands in its hydrophobic pocket, it certainly presents varying affinities depending on the nature of the ligand. The better affinity for single chain phospholipid, negatively charged,
can be explained a posteriori, by the size of the pocket and by the patch of positively charged residues located around its entrance. Further comparative studies on this protein and on homologous proteins will have to be undertaken to improve the understanding on the phospholipid binding affinity and of the phospholipid transfer activity.

Finally it remains to be explained how the structural differences observed in the plant LTP family might be related to the various in vivo activities putatively assigned to its members. Comparative structural and dynamical studies on several isoforms of wheat ns-LTPs are currently in progress in our group. This complementary work should permit us to draw stronger links between structural features and biological functions.

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REFERENCES
1. Kader, J. C., Julienne, M., and Verpoule, C. (1984) Eur. J. Biochem. 139, 411–416
2. Kader, J. C. (1996) Annu. Rev. Plant Physiol. Plant Mol. Biol. 47, 627–654
3. Kader, J. C. (1997) Trends Plant Sci. 2, 66–70
4. Sterk, P., Booij, H., Schellekens, G., Kader, J. C., and De Vries, S. C. (1994) in Plant Molecular Biology (Corruzi, G. and Puigdomenech, G. P., eds) Vol. 8, pp. 33–38, John Wiley & Sons, New York
5. Hendriks, T., Meijer, E. A., Thoma, S., Kader, J. C., and De Vries, S. C. (1994) Plant Cell 6, 1055–1058
6. Garcia-Olmedo, F., Molina, A., Segura, A., and Moreno, M. (1995) Trends Plant Sci. 1, 47–50
7. Garcia-Garrido, J. M., Menossi, M., Puig-Domenech, P., Martinez-Izquierdo, J. A., and Delseny, M. (1998) FEBS Lett. 428, 183–199
8. Krause, Á., Sigrist, C., Delling, L., Sommer, H., and Broughton, W. (1994) J. Biomol. NMR 4, 259–276
9. Kleywegt, G. J., and Jones, T. A. (1994) J. Mol. Biol. 234, 238–258
10. Kleywegt, G. J., Zou, J. Y., Kjeldgaard, M., and Jones, T. A. (1998) Acta Crystallogr. Sect. D Biol. Crystallogr. 54, 905–921
11. Laskowski, R. A., Millett, R. N., Kaptein, R., and De Dombal, J. M. (2001) Curr. Opin. Struct. Biol. 11, 269–277
12. Lehmann, B., Andersen, K. V., Nielsen, P. R., Bech, L. M., and Poulsen, F. M. (1996) Protein Sci. 5, 13–23
13. Shin, D. H., Lee, J. Y., Hwang, K. Y., Kim, K. K., and Suh, S. W. (1995) Structure 3, 189–199
14. Gomar, J., Petit, M. C., Sodano, P., Sy, D., Marion, D., Kader, J. C., Vovelle, F., and Ptak, M. (1996) Protein Sci. 5, 565–577
15. Lee, J. Y., Min, K., Cha, H., Shin, D. H., Hwang, K. Y., and Suh, S. W. (1998) J. Mol. Biol. 276, 437–448
16. Pazmanci, J., Sodano, P., Suh, S. W., Lee, J. Y., Ptak, M., and Vovelle, F. (1999) Eur. J. Biochem. 259, 692–708
17. Lehmann, B., Aagaard, R. B., Bech, L. M., and Poulsen, F. M. (1997) FEBS Lett. 416, 130–134
18. Charvolin, D., Douliez, J., Marion, D., Cohen-Addad, C., and Pebay-Peyroula, E. (1999) Eur. J. Biochem. 264, 562–568
19. Sodano, P., Caille, S., Douliez, J.-P., Marion, D., and Vovelle, F. (2000) Eur. J. Biochem. 267, 1117–1124
20. Sodano, P., Caille, S., Marion, D., Aaland, D. P., Ptak, M., Vovelle, F., and Sodano, P. (1998) Biochemistry 37, 3632–3637
21. Morin, J.-P. (1999) Characterization of a Hard Wheat Lipid Binding Protein. Purification, Sequencing and CDNA Cloning. Ph.D. thesis, Université des Sciences et Techniques du Languedoc, Montpellier, France
22. Douliez, J.-P., Patou, C., Rabesona, H., Mollé, D., and Marion, D. (2001) Eur. J. Biochem. 268, 1400–1403
23. Dharmaraj, S., You-Sen, L., Chao-Sheng, C., and Ping-Chiang, L. (2002) J. Biol. Chem. 277, 35267–35273
24. de Lamotte, F., Boze, H., Blanchard, C., Klein, C., Moulin, G., Gautier, M.-F., and Delseny, M.-A. (2001) Protein Expression Purif. 22, 318–324
25. Klein, C., de Lamotte-Gueray, F., Gautier, F., Moulin, G., Boze, H., Jourdier, P., and Gautier, M. F. (1998) Protein Expression Purif. 13, 73–82
26. de Lamotte, F., Klein, C., Issaly, N., Gautier, M.-F., and Boze, H. (1999) FEBS Lett. 457, 239–243
27. Higgins, D., Thompson, J., Gibson, T., Thompson, J. D., Higgins, D. G., and Gibson, T. J. (1994) Nucleic Acids Res. 22, 4673–4680
28. Alothul, S., Madden, T., Schaffer, A., Zhang, J., Zhang, Z., Miller, W., and Lipman, D. (1997) Nucleic Acids Res. 25, 3389–3402
29. Douguet, D., and Labesse, G. (2001) Bioinformatics 17, 732–733
30. Kelley, L. A., MacCallum, R. M., and Sternberg, M. J. E. (2000) J. Mol. Biol. 299, 501–522
31. Labesse, G., and Mornon, J. (1999) Bioinformatics 15, 206–211
32. Pons, J. L., Mallavin, T. E., and Delseny, M. (1996) J. Biol. Chem. 271, 445–452
33. Johnson, J. C. S. (1999) Prog. Nuclear Magn. Reson. Spectroscopy 34, 253–256
34. Gibb, S. J., and Johnson, C. S., Jr. (1991) J. Magn. Reson. 93, 385–402
35. Wang, D. H., Chen, A. D., and Johnson, C. S. (1995) J. Magn. Reson. Ser. A 115, 260–264
36. Humphrey, W., Dalle, A., and Schulten, K. (1996) J. Mol. Graphics 14, 33–38
37. Wuthrich, K. (1986) NMR of Proteins and Nucleic Acids, pp. 117–161, John Wiley & Sons, Inc., New York
38. Pons, J. L., and Delseny, M.-A. (1999) J. Biol. Chem. 274, 15–26
39. Mallavin, T. E., Pons, J. L., and Delseny, M. A. (1998) Bioinformatics 14, 624–631
40. Ludwig, S., Andersen, K. V., and Poulsen, F. M. (1991) J. Mol. Biol. 217, 733–736
41. Brunger, A. T., Adams, P. D., Clore, G. M., DeLano, W. L., Gros, P., Grose, C., Kuzawa, H. W., Jiang, S. J., Kuszewski, J., Nilges, M., Pannu, N. S., Read, R. J., Rice, L. M., Simonsen, T., and Warren, G. L. (1998) Acta Crystallogr. Sect. D Biol. Crystallogr. 54, 905–921
42. Kleywegt, G. J., and Jones, T. A. (1994) Curr. Opin. Struct. Biol. 4, 296–304
43. Kleywegt, G. J., and Jones, T. A. (1994) Acta Crystallogr. Sect. D Biol. Crystallogr. 50, 178–185
44. Kleywegt, G. J., and Sou, J.-Y. (2001) in Crystallography of Biological Macromolecules (Rossman, M. G., and Arnold, E., eds) Vol. F, pp. 353–366 and 366–367, Kluwer Academic Publishers, Dordrecht, The Netherlands
45. Morin, F. P., Derhuyck, W., Boudriot, Y., Jourdier, P., and Gautier, M. F. (2001) Plant Sci. 167, 747–753
46. Tannier, W., and Lehn, L. (1987) Biochim. Biophys. Acta 906, 81–99
47. de Lamotte, F., Vagner, F., Pons, J.-L., Gautier, M.-P., and Delseny, M.-A. (2001) C.R. Acad. Sci. (Paris) 3, 839–843
48. Cuff, J. A., Clamp, M. E., Squires, A. S., Finlay, M., and Barton, G. J. (1998) Bioinformatics 14, 892–893