A New Endophytic Fusarium Oxysporum Gibberellic Acid: Optimization of Production Using Combined Strategies of Experimental Designs and Potency on Tomato Growth under Stress Condition

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This study reports the potential of the endophytic fungi identified as a Fusarium oxysporum to produce gibberellic acid (GA3). The GA3 production was confirmed by high performance liquid chromatography. To improve the production of this phytohormone under solid state fermentation (SSF), successive optimization strategies were used. Firstly, Placket-Burman design was applied for screening medium components and culture condition. Under the optimized condition, GA3 yield (7.14 g/kg) was 2.62-fold higher than by the use of the initial condition (2.72 g/kg). The concentration of the most influential parameters and their interaction were optimized with a Box-Behnken experimental design. The optimized condition led to a 1.14-fold enhancement in GA3 production, reaching 8.16 g/kg. The GA3 crude extract obtained by SSF was then used to study its ameliorative role on adverse salinity effect on tomato plants (Solanum lycopersicum L.). The interactive effects of different GA3 concentrations were examined on morphological and physiological parameters of tomato plants. The application of GA3 (10⁻⁶ M) under salt stress condition (100 mM) was found to improve growth and physiological parameters including plant height, total chlorophyll, starch, and proline contents. The exogenous application of GA3 is a potent strategy to reverse abiotic stress that affect the agricultural productivity and limit plant growth and yield.

1. Introduction

Salinity is one of the major agricultural productivity limiting factors around the world. It seems to disturb the majority of plant’s metabolic processes, such as growth, photosynthesis, and biosynthesis of compatible solutes. Thus, low precipitation, surface evaporation, irrigation with poor water quality, and some agricultural practices are among the main causes of salinity. Each year, around 20 million ha of field is lost due to this phenomenon [1]. Recently, more than 800 million ha are affected. In Tunisia, this problem is faced in around 1.5 million ha [2].
ginibberellins is well known to improve abiotic stress tolerance in crops [5, 6]. For instance, gibberellins foliar spray counteracted the growth restriction caused by NaCl in wheat, rice, and tomato plants. The exogenous application of gibberellins plays a critical role in adjusting osmotic potential and sustaining plant growth under salt stress. Gibberellin acid (GA3) belongs to the family of gibberellins and it is known as an important plant growth regulator, stimulating numerous development processes. GA3 is a class of diterpenoids with a high industrial value which involves agriculture, nurseries, tea gardens, etc [7–9]. The use of this phytohormone as a plant growth regulator was extremely limited due to its scarcity and high costs (25$/g) in the market.

Generally, gibberellin industrial production is performed in submerged fermentation. However, the production cost is extremely high due primarily to elevated energy consumption and very low yield. Recently, the use of solid state fermentation (SSF) has gained a lot of attention [10]. This process simulates the biology of filamentous fungi, offering many advantages when compared to the submerged fermentation (SMF), such as higher productivity, simplicity, and lower downstream costs. Thus, SSF was considered as the best alternative to SMF techniques in the production of secondary metabolites and in the valorization of agro-resources [11, 12].

In this context, we studied the effect of a new isolated Fusarium oxysporum GA3 on the physiological, morphological, and biochemical responses of tomato plants under salt stress conditions. In this work, we have focused to optimize GA3 production under SSF on low-cost agro-industrial waste as substrates using statistical experimental designs.

2. Material and Methods

2.1. Isolation and Screening of Endophytic Fungi for GA3 Production. In order to screen GA3-producing fungi, 23 new isolates were isolated from the root of the Coriandrum sativum plant. Roots were cut into small pieces and washed with autoclaved distilled water. The samples were dipped into 4% sodium hypochlorite for 60 seconds and 70% ethanol for 5 seconds. The surface sterilized roots were washed with sterile water and left to dry under sterile conditions. The samples were inoculated on potato dextrose agar (PDA) (39 g. L−1) plates amended with 50 mg/L tetracycline to abolish the bacterial growth and incubated at 28–30°C for 3 to 4 days. The new fungi were isolated from the plates and subcultured on PDA. Subculture was continued until pure isolates were obtained. The isolates were screened for the GA3 production. The strains were harvested from the plates, dislodged under aseptic conditions, and then tested on a Czakpe medium (composed of (g/L) sucrose, 30; NaNO₃, 3; K₂HPO₄, 1; MgSO₄.7H₂O, 0.5; KCl, 0.5; and FeSO₄. 0.01 at pH 6) for 8 days at 30°C and 200 rpm. Fungi strains with potential GA3 production were retained for further study.

2.2. Molecular Identification of the Selected Fungi. Molecular biology techniques were carried out as described elsewhere [13]. The B28 strain was cultivated on a Czakpe medium at 30°C under continuous agitation at 200 rpm. Mycelia from 48-h cultures were harvested and DNA was extracted.

The internal transcribed spacer regions (ITSs) were submitted to PCR amplification using fungus-specific primers, namely ITS1 (5'-TCC GTA GGT GAA CCT GCG G-3') and ITS4 (5'-TCC TAC GTA TAT TGA TAT G-3'). The amplification was initiated by incubating the PCR reaction mixture at 95°C for 5 min, followed by 35 cycles of denaturation for 30 s at 94°C. The reaction was annealed at 50°C for 30 s and terminated with extension consisting of 1 min at 72°C and final steps of 10 min at 72°C. The PCR products were analyzed on an agarose gel (1%) and purified by the Polyethylene Glycerol- (PEG-) NaCl method. The nucleotide sequences were determined using the Big-Dye Terminator v3.1 Cycle Sequencing Kit and the automated ABI Prism 3100-Avant Genetic Analyser (Applied Biosystems). The sequence obtained for the intergenic region rRNA. The BLAST search program (http://www.ncbi.nlm.nih.gov/BLAST/) was used to look for nucleotide sequence homology.

2.3. Submerged Fermentation. F. oxysporum strain B28 culture was inoculated from the PDA slants into 250 ml of Czakpe medium (composed of (g/L) sucrose, 30; NaNO₃, 3; K₂HPO₄, 1; MgSO₄.7H₂O, 0.5; KCl, 0.5; and FeSO₄. 0.01 at pH 6) and incubated at 30°C for 10 days at 150 rpm. Separation of fungal mycelium and broth by filtration through Whatman No.1 filter paper. The filtrate was then centrifuged and the supernatant was used for GA estimation.

2.4. Estimation of Gibberellic Acid Production. GA3 was extracted and estimated by the method of Holbrook and Bailey [14]. The filtered fermented Czakpe medium (10 ml) was transferred to a centrifuge tube and added by 0.5 ml zinc acetate (1 M) solution and shaken for 3 min. This mixture was then supplemented by 0.5 ml of potassium ferrocyanide solution (1 M) and centrifuged for 15 min. 2.5 ml of supernatant was added by 0.5 ml zinc acetate (1 M) and centrifuged for 15 min. 2.5 ml of supernatant was transferred to a 250 ml flask containing 8 ml absolute ethanol and 90 ml HCl (30%). For control, 35 ml HCl solution (5%) was taken in a 250 ml flask and the volume made to 100 ml with 65 ml of distilled water. The flasks were incubated at 20 ± 2°C in a water bath for 75 min and the absorbance was read at 254 nm. The GA3 concentration was estimated using a standard GA curve. This standard curve was established from known solutions of pure gibberellic acid (obtained from Sigma Aldrich, Germany) and a linear relationship was established throughout a concentration range of 0 to 0.4 g.L⁻¹.

2.5. Extraction and Identification of Gibberellic Acid. The filtrate of the fermentation broth was acidified to pH 2.2-2.5 with 18% HCl and extracted twice with ethyl acetate. Sixty percent methanol (MeOH) with a minimum volume was added and the pH was adjusted up to 8 ± 0.3 by adding 2 N NaOH. After filtration, GA3 was added to the filtrate as an internal standard. The quantification of GAs was performed by high performance liquid chromatography (HPLC-UV, Agilent 1260
2.7. Optimization of Enzyme Production under SSF. The Plackett–Burman design, an efficient way for the screening of a large number of variables [16], was employed for choosing medium component and fermentation parameters that enhanced the GA3 production. All independent variables were tested at two levels (-) and (+), which referred to high and low, respectively. In this study, the additional nutrients were screened by a Plackett–Burman design for thirteen variables at two levels. The thirteen assigned variables were screened in 16 experimental designs. GA3 yields were taken as response (Table 1). The Taguchi methodology was used to understand the relationship between factors of medium components, adjust their concentrations, and reduce their impact on GA3 production as screened by Plackett–Burman design for thirteen variables giving the highest GA3 production. Modeling was achieved using a second-order polynomial equation:

\[ y = b_0 + \sum_{i=1}^{n} b_i X_i + \sum_{i=1}^{n} b_{ii} X_i^2 + \sum_{i=1}^{n-1} \sum_{j=i+1}^{n} b_{ij} X_i X_j, \]  

(1)

where \( y \) is the GA3 activity, \( b_0 \) is the offset term, \( b_i \) is the linear effect, \( b_{ii} \) is the squared effect, \( b_{ij} \) is the first-order interaction effect, and \( X_i \) is the independent variable.

Regression analysis of the experimental data yielded the following quadratic model equation:

\[ Y = 0.928 + 0.384 \times X_1 + 7.811 \times X_2 + 25.502 \times X_3 + 1.611 \times X_1X_2 - 0.856 \times X_1X_4 - 69.995 \times X_3X_3 - 2.749 \times X_4X_4, \]  

(2)

where \( Y \) is the GA3 production (g/kg ds) and \( X_1, X_2, X_3, \) and \( X_4 \) are, respectively, date waste, \( \text{NaNO}_3 \), urea, and \( (\text{NH}_4)_2\text{SO}_4 \).

2.8. Experimental Design. Trials were conducted at the Biotechnology Center of Sfax, Tunisia. Tomato plants with three leaves (Solanum lycopersicum L.) were transplanted into 51 pots filled with soil and plant ash (50:50). The pots were kept under ambient environment conditions with natural sunlight and temperature (from March to July, 2017). Air temperature ranged between 21 ± 2.5 and 27.5 ± 3.5°C. Relative humidity varied from 56 ± 5.5 to 70 ± 6.5.

All plants were divided into 9 groups according to the following treatments:

- **CP**: control plants untreated with NaCl and irrigated with tap water
- **SSP1**: stressed plants treated with tap water containing 50 mM of NaCl
- **SSP2**: stressed plants treated with tap water containing 100 mM of NaCl
- **SSP1+GA31**: stressed plants treated with tap water containing 50 mM of NaCl+ foliar spray application of gibberellic acid GA3 (10⁻⁶ M)
- **SSP1+GA32**: stressed plants treated with tap water containing 50 mM of NaCl+ foliar spray application of GA3 (10⁻⁶ M)
- **SSP1+GA33**: stressed plants treated with tap water containing 50 mM of NaCl+ foliar spray application of GA3 (10⁻⁷ M)
- **SSP2+GA31**: stressed plants treated with tap water containing 100 mM of NaCl+ foliar spray application of GA3 (10⁻⁶ M)
- **SSP2+GA32**: stressed plants treated with tap water containing 100 mM of NaCl+ foliar spray application of GA3 (10⁻⁷ M)
- **SSP2+GA33**: stressed plants treated with tap water containing 100 mM of NaCl+ foliar spray application of GA3 (10⁻⁸ M)

2.9. Plant Growth Measurement. At the end of each experiment, growth parameters (plant height, number of leaves per plant, and number of fruit per plant) were determined. Plant height was measured using a meter rule (cm) from the base of the plant to the apical region of the leaf. At harvest, plants were divided into roots and leaves and washed in distilled water. For the determination of fresh weight (FW), samples were dried on filter paper and weighed. Samples of leaves were either immediately used for analyses. Other samples of leaves were oven-dried at 70°C to a constant
## Table 1: Variables and nutrient screened on the Plackett-Burman design in gibberellic acid production by *F. oxysporum*.

| Run | Global Inoculum size (%) | Wheat bran (g/flask) | Sesame bark (g/flask) | Wheat straw (g/flask) | Barley bran (g/flask) | Sucrose (g/flask) | Date waste (g/flask) | Molasse (g/flask) | NaNO₃ (g/flask) | NH₄NO₃ (g/flask) | (NH₄)₂SO₄ (g/flask) | Yield (g/kg ds) |
|-----|--------------------------|----------------------|----------------------|----------------------|----------------------|------------------|--------------------|------------------|----------------|----------------|------------------|----------------|
| 1   | 1                        | 10                   | 5                    | 0                    | 0                    | 10               | 3                  | 0                | 3              | 0.5            | 0                | 0.1             | 0.5             | 3.041 |
| 2   | 1                        | 10                   | 10                   | 0                    | 0                    | 10               | 3                  | 0                | 3              | 0.5            | 0                | 0.1             | 0.5             | 1.008 |
| 3   | 1                        | 10                   | 10                   | 10                   | 0                    | 3                | 5                  | 0                | 3              | 0.5            | 0                | 0.1             | 0.5             | 2.600 |
| 4   | 1                        | 10                   | 10                   | 10                   | 0                    | 3                | 0                  | 3                | 0              | 0.5            | 0                | 0.1             | 0.5             | 2.166 |
| 5   | 1                        | 5                    | 10                   | 10                   | 0                    | 3                | 0                  | 0                | 0              | 0.5            | 0                | 0.1             | 0.5             | 2.339 |
| 6   | 1                        | 10                   | 5                    | 10                   | 10                   | 10               | 0                  | 0                | 0              | 0.5            | 0                | 0.1             | 0.5             | 1.364 |
| 7   | 1                        | 5                    | 10                   | 0                    | 10                   | 10               | 10                 | 5                | 0              | 0.1            | 0                | 0               | 0.1             | 1.483 |
| 8   | 1                        | 10                   | 5                    | 10                   | 0                    | 10               | 10                 | 5                | 3              | 0.5            | 0                | 0.1             | 0.5             | 0.830 |
| 9   | 1                        | 10                   | 10                   | 0                    | 10                   | 10               | 5                  | 3                | 0.5            | 0              | 0               | 0               | 0.5             | 2.319 |
| 10  | 1                        | 5                    | 10                   | 0                    | 10                   | 10               | 3                  | 5                | 3              | 0.5            | 0.5             | 0.1             | 0               | 1.986 |
| 11  | 1                        | 5                    | 10                   | 10                   | 0                    | 10               | 10                 | 3                | 0              | 0.5            | 0.5             | 0.1             | 0               | 1.938 |
| 12  | 1                        | 10                   | 5                    | 10                   | 0                    | 10               | 10                 | 3                | 5              | 0.5            | 0.5             | 0.1             | 0.1             | 3.009 |
| 13  | 1                        | 5                    | 10                   | 0                    | 0                    | 10               | 10                 | 3                | 0              | 0.5            | 0.1             | 0.1             | 0.5             | 1.644 |
| 14  | 1                        | 5                    | 5                    | 10                   | 0                    | 10               | 10                 | 5                | 0              | 0.5            | 0.1             | 0.1             | 0.5             | 4.274 |
| 15  | 1                        | 5                    | 5                    | 10                   | 0                    | 3                | 5                  | 3                | 0              | 0.5            | 0.1             | 0.1             | 0.5             | 1.991 |
| 16  | 1                        | 5                    | 5                    | 0                    | 0                    | 3                | 0                  | 0                | 0              | 0              | 0.5             | 0               | 0               | 1.242 |

BJ  = 1, 90 -0.03 -0.13 0.11 -0.0014 -0.11 -0.22 0.23 -0.08 0.41 -0.13 0.44 0.08 0.36
weight as described previously by Zouari et al. [17]. Finally, the oven-dried plant materials were ground in a grinder.

2.10. Chlorophyll Content. For chlorophyll a, b, and total chlorophyll analyses, fresh leaves were incubated in the dimethyl formamide in obscurity at 4°C for one week. In the end of this period, foliar pigment contents were determined spectrophotometrically according to the method of Arnon [18]. Absorbance was read at 645 nm ($A_{645}$), 652 nm ($A_{652}$), and 663 nm ($A_{663}$) using Helios β spectrophotometer (ThermoSpectronic, Courtaboeuf, France). Concentrations of chlorophyll a, chlorophyll b, and total chlorophyll were

| Variable code | Variables      | Unit      | Level (−1) | Level (+1) | b     | t value | $P > |t|$ |
|---------------|---------------|-----------|------------|------------|-------|---------|-------|
| X3            | Sesame bark   | (g/flask) | 0          | 10         | 10.968| 2.012   | 0.182 |
| X4            | Wheat straw   | (g/flask) | 0          | 10         | 9.615 | 1.763   | 0.220 |
| X7            | Date waste    | (g/flask) | 0          | 5          | 10.226| 1.876   | 0.202 |
| X9            | NaNO$_3$      | (g/flask) | 0          | 0.5        | 15.020| 2.755   | 0.101 |
| X11           | Urea          | (g/flask) | 0          | 0.1        | 12.086| 2.215   | 0.157 |
| X13           | (NH$_4$)$_2$SO$_4$ | (g/flask) | 0        | 0.5        | 10.372| 1.902   | 0.197 |

$b$: unstandardized coefficients. $P > |t|$: significance.

| Run | Sesame bark | Wheat straw | Date waste | NaNO$_3$ | Urea | (NH$_4$)$_2$SO$_4$ | Yield (g/kg ds) |
|-----|-------------|-------------|------------|----------|------|-------------------|-----------------|
| 1   | 1           | 1           | 1          | 1        | 1    | 1                 | 2.831           |
| 2   | 1           | 2           | 2          | 2        | 2    | 2                 | 3.329           |
| 3   | 1           | 3           | 3          | 3        | 3    | 3                 | 2.561           |
| 4   | 1           | 4           | 4          | 4        | 4    | 4                 | 2.519           |
| 5   | 1           | 5           | 5          | 5        | 5    | 5                 | 2.706           |
| 6   | 2           | 1           | 2          | 3        | 4    | 5                 | 2.961           |
| 7   | 2           | 2           | 3          | 4        | 5    | 1                 | 5.388           |
| 8   | 2           | 3           | 4          | 5        | 1    | 2                 | 4.184           |
| 9   | 2           | 4           | 5          | 1        | 2    | 3                 | 4.712           |
| 10  | 2           | 5           | 1          | 2        | 3    | 4                 | 2.156           |
| 11  | 3           | 1           | 3          | 5        | 2    | 4                 | 2.884           |
| 12  | 3           | 2           | 4          | 1        | 3    | 5                 | 4.055           |
| 13  | 3           | 3           | 5          | 2        | 4    | 1                 | 3.309           |
| 14  | 3           | 4           | 1          | 3        | 5    | 2                 | 5.045           |
| 15  | 3           | 5           | 2          | 4        | 1    | 3                 | 2.822           |
| 16  | 4           | 1           | 4          | 2        | 5    | 3                 | 2.166           |
| 17  | 4           | 2           | 5          | 3        | 1    | 4                 | 1.726           |
| 18  | 4           | 3           | 1          | 4        | 2    | 5                 | 2.549           |
| 19  | 4           | 4           | 2          | 5        | 3    | 1                 | 2.834           |
| 20  | 4           | 5           | 3          | 1        | 4    | 2                 | 2.710           |
| 21  | 5           | 1           | 5          | 4        | 3    | 2                 | 6.349           |
| 22  | 5           | 2           | 1          | 5        | 4    | 3                 | 2.484           |
| 23  | 5           | 3           | 2          | 1        | 5    | 4                 | 3.036           |
| 24  | 5           | 4           | 3          | 2        | 1    | 5                 | 2.851           |
| 25  | 5           | 5           | 4          | 3        | 2    | 1                 | 2.672           |
A reagent mixture was boiled at 95°C, dissolved in 100 ml of cold 95% H2SO4. After agitation, the mixture was mixed with 5 ml cold anthrone reagent (200 mg of anthrone dissolved in 10 ml of ethanol 80% in covered glass tubes and boiled at 100°C. Dry powder samples (100 mg) were mixed with 4 ml of distilled water and made up to 100 ml with distilled water. An aliquot of 1 ml of the extract was mixed with 2 ml of aqueous sulfosalicylic acid and used for estimation. The absorbance was read at 640 nm. Soluble sugar concentration was calculated using glucose solutions to develop a standard curve.

Starch contents were also determined according to the method of McCready et al. [19]. Dry powder samples (100 mg) were extracted in boiling 80% ethanol. The residue left behind after alcoholic extraction was dissolved in 5 ml of 52% perchloric acid for 1 h. The mixture was filtered and made up to 100 ml with distilled water. An aliquot of 1 ml of the extract was mixed with 4 ml of distilled water and 10 ml of cold anthrone reagent (200 mg of anthrone dissolved in 100 ml of cold 95% H2SO4). After agitation, the reagent mixture was boiled at 95°C for 10 min. After cooling, the absorbance was read at 640 nm. Soluble sugar concentration was calculated using glucose solutions to develop a standard curve. Starch concentration was calculated using glucose solutions to develop a standard curve.

2.11. Soluble Sugars and Starch Contents. Soluble sugar contents were determined according to the method of McCready et al. [19]. Dry powder samples (100 mg) were mixed with 10 ml of ethanol 80% in covered glass tubes and boiled at 70°C for 20 min. After cooling, 250 μl of the extract was mixed with 5 ml cold anthrone reagent (200 mg of anthrone dissolved in 100 ml of cold 95% H2SO4). After agitation, the reagent mixture was boiled at 95°C for 10 min. After cooling, the absorbance was read at 640 nm. Soluble sugar concentration was calculated using glucose solutions to develop a standard curve. Starch contents were also determined according to the method of McCready et al. [19]. Dry powder samples (100 mg) were extracted in boiling 80% ethanol. The residue left behind after alcoholic extraction was dissolved in 5 ml of 52% perchloric acid for 1 h. The mixture was filtered and made up to 100 ml with distilled water. An aliquot of 1 ml of the extract was mixed with 4 ml of distilled water and 10 ml of cold anthrone reagent (200 mg of anthrone dissolved in 100 ml of cold 95% H2SO4) and boiled at 100°C for 10 min. After cooling, the absorbance was read at 630 nm. Starch concentration was calculated using glucose solutions to develop a standard curve.

2.12. Proline Content. Proline content was determined according to Bates et al. [20]. One gram of fresh leaves was homogenized with a mortar and pestle with 10 ml of 3% aqueous sulfoisalicylic acid. The homogenate was filtered through Whatman No 1 filter paper and the residue was reextracted. The extracts were pooled and made up to 10 ml with aqueous sulfoisalicylic acid and used for estimation. An aliquot of 2 ml of the extract was mixed with 2 ml of glacial acetic acid and 2 ml ninhydrin acid and boiled at 100°C for 1 h. After cooling, 2 ml of toluene were added to the mixture. The chromophore-containing toluene was separated, and the absorbance was measured at 520 nm with a UV/vis spectrophotometer. Toluene was used as a blank, and proline content was calculated using L-proline for the standard curve.

2.13. Ion Concentrations. The leaves used for ion determinations were washed with distilled water, oven dried at 70°C for 72 h, and ground to a fine powder. One gram of the powder was placed at 250°C for 3 h in an oven and then transferred to 100 ml of dilute nitric acid (1 M). Na+, K+, and Ca2+ concentrations were determined using a flame photometer (Jenway, PEP-7). For chloride determination, dry plant material was extracted with 40 ml HNO3 (0.2 N).

2.14. Statistical Analysis. One-way analysis of variance (ANOVA) was used. All analyses were done using the SPSS program (V23.0) by Tukey’s post hoc test to determine the significant differences between treatments. The results were expressed as mean values and standard errors from the three replications. All tests were performed at a 0.05 level of significance.

3. Results and Discussions

3.1. Isolation, Screening, and Identification of GA3-Producing Fungi Strain. Maximum GA3 production was achieved by the strain B28 (0.371 g/l). Consequently, this fungal strain was selected for further studies as the most potent isolate for GA3 production. The culture filtrate of fungal strain was subjected to chromatography analysis for the determination of GA3. The results showed that the culture filtrate of strain B28 contained GA3 (Figures 1(a) and 1(b)). The ITS amplification of the strain by PCR resulted in a product of 528 bp in size. The PCR-amplified ITS sequence from B28 was determined (GenBank accession no. MN816007) and was 99% similar to that of Fusarium oxysporum. Therefore, the strain was identified as B28 strain Fusarium oxysporum.

3.2. Optimization of GA3 Production

3.2.1. Screening of the Main GA3 Production Factors. The obtained result (Table 1) showed a wide variation of GA3 production ranging from 0.89 g/kg ds to 4.27 g/kg ds. This variation illustrates the importance of this step to screen the most influential variable and the level of the others [13]. The analysis of the contrast coefficient (b) indicated that sesame bark, wheat straw, date waste, NaNO3, urea, and (NH4)2SO4 displayed a positive effect on GA3 production; whereas inoculums size, sucrose, molasses, and fish meal...
had a negative effect on it. The critical effect of these variables on GA3 production was confirmed by statistical analyses, particularly by t test and P value (Table 2). The variables having the most important contrast coefficients including NaNO₃, urea, sesame bark, (NH₄)₂SO₄, and date waste were the most significant variables affecting GA3 production by statistical analysis. These factors presented a very low P value and the highest level of significance with a t value of 2.75, 2.21, 2.012, 1.902, and 1.87, respectively, demonstrating their pronounced effect on GA3 production. The present study is the first contribution towards the use of that sesame bark, wheat straw, and date waste, which are agro-wastes and readily available complex substrates as a complex organic source for the production of GA3 by B28 strain Fusarium oxysporum.

3.3. Optimization of GA3 Production by the Taguchi Method.

The Taguchi method was successfully used to examine the effects of different medium components such as carbon and nitrogen sources influencing the production of various microbiology metabolites [21–24].

In this study, the L25 Taguchi design was used to determine the relative effect of the selected factors (sesame bark, wheat straw, date waste, NaNO₃, urea, and (NH₄)₂SO₄ on GA3 production by B28 strain Fusarium oxysporum and also to choose the most adequate substrate among the studied ones. The analysis of the results showed that the GA3 production strictly depends on the nutritional medium components corresponding to the combined effect of the examined parameters over their defined ranges. Indeed, a huge variation in production yield ranging from 1.72 g/kg of substrate (run 17) to 6.35 g/kg of substrate (run 21) was observed (Table 3). This variation revealed the relevance of this approach to choose the best influential variable and the level of each factor. Table 4 shows that GA3 production is mainly influenced by the concentration of (NH₄)₂SO₄ which represents the largest contribution compared with the other factors (29%). This critical factor causes the greatest response in GA3 production at level 2. Sesame bark and NaNO₃ also presented an important contribution of about 17 and 18%, respectively. Maximum response occurred at level 2 and level 4, respectively. The results also showed that GA3 production in SSF is evenly influenced by the other tested variables (wheat straw, urea, and date waste) which represent significant contributions greater than 10%. Wheat straw has a larger effect on the GA3 production yield in level 4 whereas urea and date waste have the greatest effect at level 5. This result proves the importance of our optimization strategy. Indeed, the Taguchi design was successfully applied to determine the relative importance of the preselected factors and to select the suitable solid substrate for GA3 production in solid state fermentation. It is well documented that solid state fermentation processes are largely influenced by the nature of the solid substrate. The substrates are generally water insoluble polymers of cellulosic or starchy material [25]. Our study indicated that GA3 production was increased by the addition of different compounds as inducers to the media. Therefore, GA3 production increased by the use of date waste and wheat straw at its higher levels. Several studies have reported the production of GA3 in SSF using waste and byproducts [9, 10, 24]. Panchal and Desai [26], who investigated the production of GA3 of Fusarium moniliforme in SSF, used the commercial wheat bran (CWB) mineral salt acid as a solid substrate. They reported that the increase of GA3 production may be in direct correlation with the complexity of the carbon sources. Citric pulp was largely used as substrate/support for GA3 production in SSF [24, 27–29]. Nevertheless, there were no earlier reports using date waste and wheat straw as carbon source and solid substrate for GA3 production in SSF. Thus, this study may be the first report to note that date waste and wheat straw were potential substrates for GA3 production. It was interesting to note that soluble cellulose and hemicellulose fractions of wheat byproduct served as potent carbon sources which led to a sufficient carbon and nitrogen ratio for efficient metabolites production [21, 23, 30]. Date waste was investigated as an excellent growth substrate for the production of glucose oxidase in SSF [23].

The validity of the optimum conditions (7.5 g/flask sesame bark, 12.5 g/flask wheat straw, 12.5 g/flask dates waste, 1 g/flask NaNO₃, 0.5 g/flask urea, and 0.5 g/flask NH₄NO₃) showed an enhanced GA3 yield of 7.14 g/kg ds, which was 2.63-fold higher than at the initial conditions (2.72 g/kg ds).
3.4. Statistical Optimization of Gibberellic Acid Production by Box-Behnken Design. Based on the results of Taguchi methods, four factors (date waste, NaNO₃, urea, and (NH₄)₂SO₄) were found to have a greater influence on GA₃ production by the new isolate strain B28. Afterwards, the Box-Behnken design was used to optimize the level of each of these factors giving the highest production and to study their interaction. Experimental conditions and results for GA₃ yield are presented in Table 5.

The calculated regression analysis indicated that the $F$ value was 9.745, with a very low probability value ($P < 0.0001$) showing the significance of the model. The closeness of experimental and predicted GA₃ production can be expressed by the determination coefficient ($R^2 = 0.78$) which stipulates that only 22% of the total variation could not be explained by the model. Adjusted R Square of 0.70 indicated that the regression model could be used to analyze trends of responses. The analyses of the quadratic model showed that urea present the largest effect on GA₃ production in SSF. This variable shows a significant positive linear effect ($X_3$) and an important negative quadratic term ($X_3X_3$). The linear and the quadratic terms of the other variables as well as the first-order interaction between the tested factors did not show any major significant effect. These results were confirmed by Student’s $t$ test ($\alpha = 0.05$) (Table 6). The chosen model equation showed that the predictive model was selected based on the highest Adjusted R Square. Also, we noticed that the linear term of (NH₄)₂SO₄ is eliminated from this model, which means that this factor did not have a first-order significant effect on the GA₃ production. These results are in good agreement with the studies done by Panchal and Desai [26] who tested different nitrogen sources (NH₄Cl, NH₄NO₃, (NH₄)₂SO₄, (NH₄)₂MoO₄ and urea), as an additional substrate of commercial wheat bran, on the Fusarium moniliforme GA₃ production. They found that urea had a remarkable effect on GA₃ production and demonstrated that urea exhibits buffering activity and thus resists the change in pH during fermentation. The authors also noted that the lower production levels of GA₃ with ammonium salts may be due to the decrease in pH with the utilization of NH₄ ions.

The data indicated that carbon and nitrogen sources can act as nutrients or limiting factors. In fact, a slight change in their concentrations can lead to a significant variation in the production level. An increase that reached up to 7.95 g/flask in the GA₃ production in SSF was recorded with the increase of NaNO₃ concentration to 0.75 g/flask and the decrease of date waste concentration to 5 g/flask. Production levels (6.37 g/flask) were noted to increase with the decrease of (NH₄)₂SO₄ and date waste concentrations. Rodrigues et al. [30] reported that the quality and quantity of nitrogen sources are very important factors for the gibberelin production due to the ammonium regulation of the process. They noted that the higher GA₃ yields are reached when the nitrogen concentration is low in the media. The GA₃ production begins when nitrogen is exhausted [31, 32].

Under the optimized culture media, the quadratic model showed that the maximum GA₃ production would be 8.29 g/flask, when date waste, NaNO₃, urea, and (NH₄)₂SO₄ were 5, 0.75, 0.19, and 0.25 g/flask, respectively. To validate the predicted results, fermentation experiments were performed in two tests. GA₃ production yield (8.16 g/flask) was absolutely more important than that obtained during the initial study (2.72 g/flask). Thus, the production level was multiplied by a factor of 3. In this study, the successive optimization strategies (Plackett-Burman, L25 Taguchi method, and Box-Behnken design) were successfully applied to test the relative importance of medium components in the GA₃ production. The finding indicates that, after optimization of the culture medium, the production levels of GA₃ by Fusarium oxysporum were reached before (Table 7).

3.5. Effects of GA₃ Exogenous Application on Tomato. In the current experiments, salt stress conditions induced a reduction ($P < 0.05$) in morphological traits such as the plant height, number of leaves and fruits per plant, and fresh and dry plant weight (Table 8). The reduction in the numbers of leaves and fruits per plant in salt condition was, respectively, 33.6 and 45.8% in SSP1 and 46.9 and 66.6% in SSP2. The inhibition of growth characteristics of tomato plants were significantly alleviated ($P < 0.05$) by the GA₃ foliar spray application. In fact, the exogenous application of GA₃
using different concentrations (10^{-5}, 10^{-6}, \text{ and } 10^{-7} \text{M}) increased shoot length, leaf and fruit numbers, and fresh and dry shoot weights. Interestingly, maximum values of growth traits were obtained with the GA3 level of 10^{-6} \text{M} (GA32). The SSP1+GA32 and SSP2+GA32 treated plants showed an improvement (P < 0.05) by 1.67 and 1.32 times, respectively, when compared to the untreated plants with GA3 (SSP1 and SSP2). The utilization of GA3 crude extract obtained by SSF to ameliorate the adverse salinity effect on tomato plants (Solanum lycopersicum L.) is being reported for the first time. Several reports showed that the exogenous application of GA3 alleviates the adverse effects of salinity stress on plant growth [33, 34]. Maggio et al. [35] reported that in tomato plants, exogenous application of GA3 increased morphological traits and improved the yield.

Under saline conditions, toxic ions such as chloride and sodium were accumulated in the tissue of the majority of plants [36, 37]. In fact, osmotic adjustment is completed by ion uptake or through the accumulation of compatible solutes. The effect of salt stress on macronutrient content of tomato plants is presented in Figures 2(a) and 2(b). The finding revealed that in SSP1 and SSP2 treated plants, salinity was shown to increase the Na+ and Cl- levels and to reduce the K+ and Ca2+ levels, however, exogenous application of GA3 resulted in a significant improvement of Ca2+ and K+ levels, whereas, exogenous application of GA3 showed a significant improvement of Ca2+ and K+ contents (Figure 2(a)) and an important reduction (P < 0.05) of Na+ and Cl- contents (Figure 2(b)) in the leaves of tomato plants. As compared to the SSP2-treated plants, the K+ and Ca2+ contents of SSP2+GA32-treated plants were, respectively, increased by 42 and 36%, and the contents of Na+ and Cl- were reduced by 44 and 45%, respectively. These

### Table 6: Analysis of the model terms by Student’s t test.

| Coefficients | Nonstandardized coefficients | Standardized coefficients | t | Significance |
|--------------|-------------------------------|---------------------------|---|-------------|
|              | B                             | Beta                      |    |             |
| X1           | 2.004                         | 0.931                     | 0.00 | 2.153       | 0.044 |
| X2           | 0.261                         | 0.093                     | 1.410 | 2.823       | 0.011 |
| X3           | 6.847                         | 1.437                     | 1.801 | 4.763       | 0.000 |
| X1X2         | 26.871                        | 5.549                     | 2.900 | 4.843       | 0.000 |
| X1X4         | -0.735                        | 0.138                     | -2.829 | -5.347   | 0.000 |
| X3X3         | 0.146                         | 0.118                     | 0.569 | 1.243       | 0.229 |
| X4X4         | -69.624                       | 13.649                    | -3.057 | -5.101     | 0.000 |

### Table 7: GA3 production by solid state fermentation in the literature.

| Substrates                  | Nitrogen source         | Fermentation technique | Production   | Reference |
|-----------------------------|-------------------------|------------------------|--------------|-----------|
| Coffee husk cassava bagasse | (NH4)2SO4               | SSF                    | 492.5 mg of GA3/kg | Machado et al. [27] |
| Citric pulp                 |                         | SSF                    | 7.8 g/kg of dry substrate | Satpute et al. [29] |
| Pigeon pea pod              | NH4NO3                  | SSF                    | 1160 μg/g of dry substrate | Rodrigues et al. [28] |
| Commercial wheat bran       | NH4Cl, NH4NO3, (NH4)2SO4, (NH4)MoO4 | SSF                    | 160 μg/g of dry substrate | Rodrigues et al. [52] |
| Citric pulp and soy husk    | (NH4) NO3, (NH4)2 SO4, urea | SSF                    | 7.0 g/kg of dry substrate | Oliveira et al. [24] |
| Citric pulp                 |                         | SSF                    | 7.60 g/kg of dry substrate | Rodrigues et al. [52] |

### Table 8: The effect of saline condition and exogenous gibberellic acid on growth parameters of tomato plants.

| Treatments | Plant height (cm) | Leaf number per plant | Fruit number per plant | Fresh material (g/plant) | Dry material (g/plant) |
|------------|------------------|-----------------------|------------------------|-------------------------|------------------------|
| CP         | 69.6 ± 3.05a     | 108 ± 6.08d           | 8 ± 1d                 | 76.5 ± 2.33e            | 11.03 ± 1.06c           |
| SSP1       | 46 ± 1.74b       | 71.66 ± 3.51b         | 4.33 ± 0.57abc         | 51 ± 1.56abc            | 7.46 ± 0.15bc           |
| SSP2       | 36.43 ± 0.81a    | 57.33 ± 1.52a         | 2.66 ± 0.57a           | 45.3 ± 2.53a            | 6.0 ± 0.17a             |
| SSP1+GA31  | 57.7 ± 2.68c     | 90.33 ± 5.50c         | 6 ± 1c                 | 55.93 ± 1.48bc          | 8.1 ± 0.17c             |
| SSP1+GA32  | 62.86 ± 3.47c    | 98.33 ± 6.11d         | 6.33 ± 0.57c           | 66.46 ± 3.81d           | 9.56 ± 0.20d            |
| SSP1+GA33  | 58 ± 1.11c       | 90 ± 1c               | 5 ± 1bc                | 59.36 ± 1.09c           | 8.36 ± 0.37c            |
| SSP2+GA31  | 42.9 ± 1.9b      | 67.33 ± 3.21ab        | 2.66 ± 0.52a           | 54.16 ± 0.76bc          | 7.56 ± 0.15bc           |
| SSP2+GA32  | 48.1 ± 1.85b     | 73 ± 2.64b            | 3.33 ± 0.57ab          | 59.86 ± 3.47c           | 8.16 ± 0.15c            |
| SSP2+GA33  | 37.73 ± 2.05b    | 59 ± 5.29b            | 3 ± 1ab                | 44.8 ± 0.45a            | 6.66 ± 0.20ab           |

± Standard error. Different letters in the same column are significantly different (P < 0.05).
results agree with Tuna et al. [38], who reported that GA3 reduced the accumulation of Na⁺ and enhanced K⁺ and Ca²⁺ levels in leaves of maize plants under saline conditions. The characteristics of tomato plants grown under salt stress were effectively improved by the GA3 application. It was indicated that Gibberellic acid was useful to enhance ion contents of many plants under stress conditions, such as tomato, wheat, and maize [35, 39, 40].

In this study, we also determined the proline contents in plants untreated and treated with GA3 under salt stress conditions (Figure 3). The data showed that proline content increased in leaves with the increment of salt stress. In fact, in SSP2-treated plants, proline content was about 0.44 mg/g DM in leaf tissues. The accumulation of proline, an amino acid protectant, was reported in olive plants (Olea europaea) exposed to salt stress [41, 42]. These reports suggested that proline is also a metabolite of adaptation to the salinity. It was reviewed that under salinity stress, high levels of proline in leaves of various plants could be explained by gene expressions encoding enzymes of proline synthesis (pyrroline-5-carboxylate) or by a decrease in proline enzymes such as oxidation proline dehydrogenase [43]. Indeed, proline accumulation in plant tissues is an adaptive strategy in stressful environments, which maintains the osmotic balance, stabilizes cell membrane structure, and regulates cellular redox potential [6, 44–46]. The obtained results (Figure 3) also revealed that exogenous F. oxysporum GA3 application decreased (P < 0.05) proline content under different salinity treatments. In fact, in SSP1+GA32-treated plants, this decrease was about 35% in leaves of tomato plants in comparison to the values recorded in SSP1-treated plants (0.4 mg/g DM). These results were in agreement with the study of Tuna et al. [38], who noted that the exogenous application of GA3 alleviates the adverse effect of NaCl by decreasing the proline content in maize plants.
Solutes, such as sugars, accumulate in the cytosol under salt stress and can therefore contribute to plant survival. They play an important role in the osmoregulation under several conditions [47]. In this study, leaves of tomato plants exposed to two salt stress treatments accumulated high soluble sugar contents ($P < 0.05$) as compared to the unstressed plants (CP). It was about 41.1 (mol/g DM) in SSP1 and 38.5 (mol/g DM) in SSP2. However, the exogenous application of $F. oxysporum$ GA3 at both levels led to the decrease of soluble sugars contents ($P < 0.05$) in leaf tissues of tomato plants (Figure 4). Indeed, the reduction of soluble sugars contents, in the presence of $10^{-6}$ M GA3, was 1.6 and 1.2-fold less than SSP1 and SSP2-treated plants, respectively. These results are in accordance with those obtained by Iqbal et al. [43].

Data presented in Figure 5 show that starch content in leaves of tomato plants were significantly reduced ($P < 0.05$) by both salinity treatments, when compared to the unstressed control plants. The lowest values of starch contents were recorded in SSP2-treated plants (0.67 (mol/g DM)). The externally applied GA3 increased the starch content in treated plants, but the levels were still lower than those of control

![Figure 3: The effect of salt stress and exogenous application of GA3 on proline content in tomato leaves. The different letters (A, B, C, D, and E) indicate significant differences among treatments ($P < 0.05$) according to Tukey’s test.](image)

![Figure 4: Accumulation of soluble sugar content in leaves of tomato plants under saline conditions and exogenous gibberellic acid. The different letters (A, B, C, D, and E) indicate significant differences among treatments ($P < 0.05$) according to Tukey’s test.](image)
plants (1.65 (mol/g DM)). Under stressed conditions, the highest amounts of starch content in leaf tissues were recorded in SSP1+GA32-treated plants.

The higher accumulation of sugars and proline in leaves (\(P<0.05\)), compared to unstressed plants, could be due to their photosynthetic activity. In fact, leaves of tomato plants with low photosynthetic activity tend to synthesize more soluble sugars. Furthermore, the decrease in starch content in plants under stressed conditions could be due to starch degradation and/or to the increment in soluble sugar concentration under limited water availability. Todaka et al. [48] demonstrated that in stressed plants, \(\beta\)-amylase activity increased. The lower starch concentrations in leaves of salt-stressed plants suggest that carbon was translocated out of the leaves.

Salt stress disturbs several aspects of plant mechanisms, such as photosynthesis and pigment synthesis [49]. The data in Figure 6 show that 50 and 100 mM NaCl treatments caused a significant decrease (\(P<0.05\)) of chlorophyll (a, b, and a+b) content, in comparison to control plants. The GA3 treatment increased (\(P<0.05\)) the contents of photosynthetic pigments (Figure 6). The highest levels were recorded in SSP1+GA32-treated plants. These values were 0.82, 0.21, and 1.06 mg/g of FM, for chlorophyll (a, b, and a+b) contents, respectively. In the case of soybean plants, Zhao et al. [50] reported that treatment with GA3 increased pigments content. Also, it has been demonstrated that chlorophyll and carotenoids contents of maize leaves were increased by the treatment with GA3 [51].

4. Conclusion

The endophytic fungus \(F. \text{ oxysporum}\), isolated from root of flowers plant, showed a good ability to produce maximal yield of gibberellic acid on low-cost substrate under SSF. Under the optimization, the maximal yield of GA3 (8.16 g/kg ds) increased about 2.72-folds from the initial production medium. In the present work, the application of GA3 at \(10^{-6}\) M provided an efficient regulator for tomato plants under salt stress.

**Abbreviations**

\((\text{NH}_4)_2\text{SO}_4\): Ammonium sulfate  
CW:B: Commercial wheat bran  
DM: Dry matter  
FeSO_4: Ferrous sulfate
FM: Fresh matter
GA3: Gibberelic acid
H2SO4: Sulfuric acid
HCL: Chlorhydric acid
HNO3: Nitric acid
ITS: Internal transcribed spacer
K2HPO4: Dipotassium phosphate
KCl: Potassium chloride
MeOH: Methanol
MgSO4·7H2O: Magnesium sulfate heptahydrate
NaCl: Sodium chloride
NaNO3: Sodium nitrate
NH4NO3: Ammonium nitrate
PCR: Polymerase chain reaction
PDA: Potato dextrose agar
SMF: Submerged fermentation
SSF: Solid state fermentation.

Data Availability

The datasets generated and/or analyzed during the current study are available on the GenBank repository, https://www.ncbi.nlm.nih.gov/genbank/. The GenBank accession number for the nucleotide sequence of the 5.8S its gene referred to in the text is MN816007. Other datasets generated during and/or analyzed during the current study are available from the corresponding author on reasonable request.

Conflicts of Interest

The authors declare that they have no conflicts of interest.

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