EFFECT OF THE VARIABILITY OF HEAT AND MASS TRANSFER COEFFICIENTS ON 3D UNSATURATED POROUS MEDIUM DRYING

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Abstract

A three-dimensional unstructured control volume finite method is developed to simulate unsteady coupled heat and mass transfer phenomena that arise during convective drying of unsaturated porous media. In order to simulate 3D geometries, as application here the drying of clay brick portion, we developed a Fortran code based on 3D unstructured meshes generated by the free mesh generator Gmsh. Several simulation results are presented and depict the effect of the variability of heat and mass transfer coefficients. These simulations prove that only three-dimensional model is able to capture the effect of variability of heat and mass transfer coefficients.

Keywords: Porous medium, 3D drying, Gmsh, MVCEF, mass and heat transfer

1. INTRODUCTION

The high-energy consumption during porous solids drying has made it an interesting subject in a great number of industrial applications, including wood, food, textile, ceramics, building materials, etc. (Caccavale et al., 2016; Almeida et al., 2013). Since, to reduce the drying time hence the energy consumption, we should simultaneously, improve both the circulation of water and gas-diffusion and accelerate surface evaporation. Under specific conditions, convective drying ensures not only the reduction of drying time but also preservation of the quality of the products at the same time. Therefore, heat and mass transfer should be carefully investigated throughout drying process.

However, the high cost and the time involved in experimental studies have made the numerical modeling widely used in analyzing the drying phenomena. Consequently, considerable researches have been conducted to numerically simulate the drying process (Defraeye, 2014; Liu et al., 2015; Sadrmontazi and Haghi, 2008).

Drying is a process involves a coupled heat and mass transfer in a multi-phase flow in porous media. Based on the theory proposed by drying Luikov (Luikov, 1975) and later by Whitaker (Whitaker, 1977), several mathematical models for drying porous materials have been developed. The equations that govern the transport processes associated with flow in porous media are tightly coupled and highly non-linear. The discretization of these equations leads to a system of three partial differential equations for dynamic content variables as humidity, temperature and pressure. Several comprehensive theoretical two-dimensional models are available for studying the drying process (Ben Khedher and Ben Nasrallah, 2010; Helel and Boukadida, 2008).

A review of the literature unveils little evidence of where unstructured mesh control volume method has been used to study numerically the heat and mass transfer in porous materials. Ferguson et al. (Ferguson and Turner, 1995a) have carried out a comparison of the finite element and control volume solution techniques applied to timber drying problems. Recently, structured mesh control volume techniques have been used to study the drying of wood (Ferguson and Turner, 1995b). Turner and Ferguson (Turner and Ferguson, 1995a) have used the finite element method to study similar problems. However, they found that the CVFEM was superior for reasons concerning efficiency, conservation, and robustness across a wide range of varying drying conditions of isotropic and anisotropic porous material. More recently, three-dimensional homogeneous wood drying models have been developed. In (Dedic et al., 2003), a simplified model based on the conservation of mass and energy using constant parameters obtained by comparing experimental data and numerical results for the drying of a block of beech wood was reported. Similarly, Perré and Turner (Perré and Turner, 1999), have developed a 3-D version of TransPore explaining a mass transfer model to simulate the drying of porous media. They showed that only three spatial dimensions are able to capture the correct physical behavior of the porous medium throughout the drying process and that the thickness and width of the board are necessary to account for the coupled mechanisms of thermal and mass diffusion, and Darcy’s flow.

To date a very few of comprehensive theoretical three-dimensional models are available for studying the drying process. In previous works we developed a three dimensional tool for simulating the drying of a whole brick under industrial drying conditions (Rzig et al., 2017). However, in the previous studies, the heat transfer coefficient was considered constant and the mass transfer coefficient was deduced by assuming analogy between heat and mass transfer. These two hypothesis lead to information deficiency as far as the real mechanism of heat and mass transfer during convective drying of porous media is concerned.

In this context, this study focuses on the effects of the variability of heat and mass transfer coefficients with surrounding air on the drying of clay brick. Consequently, the 3-D drying kinetics (average moisture content, temperature and pressure profiles) will be presented and analyzed.
2. MATHEMATICAL METHOD

Instead of simulating the drying of an entire brick (brick with 12 holes) and taking into account the symmetry of the problem, we will simulate heat and mass transfer in a single cavity. This square cavity is made of brick of length $l = 30$ cm and of width $L = 5.5$ cm. Only the internal cavity walls are exchanging heat and mass (Fig. 1).

Fig. 1 porous brick cavity.

2.1 Governing equations

Based on Whitaker theory, a mathematical model governing heat and mass transfer is established for the unsaturated porous media.

In order to obtain a closed set of governing macroscopic equations, the following assumptions are made:

- The porous layer is homogenous and isotropic.
- The solid, liquid and gas phases are in local thermodynamic equilibrium.
- The compression-work and viscous dissipation are negligible.
- The gas-phase is ideal in the thermodynamic sense.
- The dispersion and tortuosity terms are interpreted as diffusion term.
- The radiative heat transfer is negligible.

Considering these assumptions, macroscopic equations governing heat and mass transfer in the porous medium are:

- **Generalized Darcy’s Law:**

  The average velocities of $\bar{V}_l$ (the liquid phase) and $\bar{V}_g$ (the gas phase) are obtained using Darcy’s Law which is generalized by using the concept of relative permeability defined as the ratio between the effective permeability and the intrinsic permeability.

  - For the liquid phase:
    $$\bar{V}_l = -\frac{KK_l}{\mu_l} \left[ \nabla (P_g^l - P_c) - \frac{1}{\rho_l} \bar{\nabla} \rho_l \right]$$  \hspace{1cm} (1)

  With $P_c = P_g^l - P_l^l$ : the capillary pressure.

  - For the gas phase (without taking into consideration the gravitational effect):
    $$\bar{V}_g = -\frac{KK_g}{\mu_g} \nabla P_g$$ \hspace{1cm} (2)

  **Mass conservation equations:**

  - Liquid phase:
    Assuming that liquid density is constant, the mass conservation equation of the liquid phase is:
    $$\frac{\partial \rho_g}{\partial t} + \nabla (\bar{\rho}_g \bar{V}_g) = -\bar{m}_g$$ \hspace{1cm} (3)

    Where $\bar{m}_g$ is the mass rate of evaporation and $\bar{\varepsilon}_g$ is the volume fraction of liquid phase.

  - Gas phase:
    For this phase the average density is not constant. In this case, the mass conservation equation of the gas phase is given by:
    $$\frac{d \bar{\rho}_g}{dt} + \nabla (\bar{\rho}_g \bar{V}_g) = \bar{m}_g$$ \hspace{1cm} (4)

    Where $\bar{\rho}_g$ is the intrinsic average density of the gas phase. This phase is considered as an ideal mixture of perfect gases.

    - Vapor phase:
      $$\frac{d \bar{\rho}_v}{dt} + \nabla (\bar{\rho}_v \bar{V}_v) = \bar{m}_v$$ \hspace{1cm} (5)

    - Gas phase:
      $$\bar{\rho}_g \bar{V}_v = \rho_g \bar{V}_g - \rho_g \bar{D}_{eff} \left( \frac{\bar{P}_v}{\bar{\rho}_g} \right)$$ \hspace{1cm} (6)

    $\bar{D}_{eff}$ represents the coefficient of the effective diffusion of the vapor in the porous medium. This coefficient takes into account the resistance to the diffusion due to tortuosity and the effects of constriction.

    - Energy conservation equation:
      $$\frac{\partial}{\partial t} \left[ \bar{\rho}_v \bar{C}_p T \right] + \nabla \left[ \bar{\rho}_v \bar{C}_p \bar{V}_v + \sum_{k=a,\nu} \bar{\rho}_k \bar{C}_k \bar{V}_k \right] = \bar{\nabla} \cdot \bar{\lambda}_{eff} \bar{\nabla} T - \Delta H_{vap} \bar{m}_v$$ \hspace{1cm} (7)

    $\Delta H_{vap}$ is the latent heat of vaporization at temperature $T$(K).

    $\lambda_{eff}$ and $\bar{\rho}C_p$ are respectively the effective thermal conductivity and the constant pressure heat capacity of the porous medium and $\bar{\rho}C_p$ is given by:
    $$\bar{\rho}C_p = \rho_s C_{ps} + \rho_l C_{pl} + \rho_a C_{pa} + \rho_v C_{pv}$$ \hspace{1cm} (8)

    Where $\rho_s C_{ps}$, $\rho_l C_{pl}$, $\rho_v C_{pv}$ and $\rho_a C_{pa}$ are respectively the mass heat capacities of the brick solid matrix, liquid, vapor and air.

    - Thermodynamic relations:
      The partial pressure of the vapor is equal to its equilibrium pressure:
      $$P_v = P_{eq}(T, S)$$

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Fig. 1 porous brick cavity.
Where \( S \) is the liquid saturation defined by:

\[
S = \frac{\xi_1}{\epsilon}
\]  

(9)

The gaseous phase is assumed to be an ideal mixture of perfect gases:

\[
\bar{P}_i = \bar{P}_g, \quad \bar{P}_i = \frac{P_i}{M_i} + \rho_{vi} \quad \bar{P}_g = \bar{P}_g + \bar{P}_v
\]  

(10)

\[
\frac{P_v}{P_{vs}} = \exp \left( \frac{2.\sigma_i M_i}{r_i \rho_i RT} \right)
\]  

(11)

### 2.2 Boundary conditions

Initially, the temperature, the gas pressure and liquid saturation are uniform in the brick.

Thermal energy brought by air convection is necessary for water evaporation and to the heat conduction in porous medium. This energy is a function of temperature and heat transfer coefficient (Fig. 3).

On the exchanging faces (internal faces of the cavity), we can write:

\[
\lambda_{eff} \left( \frac{\partial T}{\partial x} \right) + \Delta H_{vap} \rho_i \left( \langle v \rangle_i \right) n_i = h_{ix} \left( T - T_{\infty} \right)
\]  

(12)

The mass flow corresponding to evaporation and to the evacuation of water is a function of the vapor density difference and mass transfer coefficient:

\[
\left[ \rho_i \langle v_i \rangle + \langle \rho_i \rangle \langle v \rangle \right] n_i = h_{ix} \left( C_{vs} - C_{v,\infty} \right)
\]  

(13)

The gas pressure on exchanging face is equal to atmospheric pressure:

\[
\left[ \langle P_g \rangle \right] = P_{atm}
\]  

(14)

Because of the symmetry of the problem the external faces of the cavity are kept impermeable.

\[
\lambda_{eff} \left( \frac{\partial T}{\partial x} \right) + \Delta H_{vap} \rho_i \left( \langle v_i \rangle \right) n_i = 0
\]  

(15)

\[
\left[ \rho_i \langle v_i \rangle + \langle \rho_i \rangle \langle v \rangle \right] n_i = 0
\]  

(16)

\[
\left[ \frac{\partial P_g}{\partial x} \right] = 0
\]  

(17)

In this study, we will test the variability of heat and mass transfer coefficients on the drying phenomena. The convective heat and mass transfer coefficients were listed in Table 1:

### Table 1 Convective heat and mass transfer coefficients.

| \( h_i \) | \( \frac{\lambda \times 0.023 \times R^{0.5} \times P_i^{0.5}}{L_i} \) |
| \( h_m \) | \( \frac{D_{A,B} \times 0.023 \times R^{0.5} \times S^{0.5}}{L_i} \) |
| Validity | \( R_s < 5 	imes 10^5 \) |
| \( P_i \geq 0.6 \) | \( S_i \geq 0.6 \) |
| \( P_i = \frac{C_m \mu_s}{\lambda_i} \) | \( S_i = \frac{\rho_{vi}}{D_{A,B}} \) |

With:

- \( h_i \): the convective heat transfer coefficient \( (W/m^2°C) \)
- \( h_m \): the convective mass transfer coefficient \( (m^3/kg) \)
- \( D_{A,B} \) is the vapor diffusion in air given by:

\[
D_{A,B} = D_{vap,atm} = 0.26 \times 10^{-4} \quad (m^2/s)
\]  

(18)

- \( L_i \) is the characteristic length of brick.

### 3. SOLUTION METHOD

The equations set, with initial and boundary conditions has been solved numerically using the Control Volume Finite Element Method (CVFEM). The advantages of this method are (i) It ensures the flux conservation (ii) the used control volumes present more faces, which makes it possible to avoid the numerical diffusion. (iii) The control volume is composed of triangular elements that improve the grid flexibility (Figure 2). For additional details the reader can refer to (Rzig et al., 2017).

#### 3.1 Mesh Generation

For the mesh generation (Fig. 2), the free mesh generator Gmsh was employed (Geuzaine and Remacle, 2009). The brick domain is divided first in six-node prisms. To create the polygonal control volumes around each node in the finite element grid, the centroids of the triangular elements (bases of prisms) are joined to the midpoints of the corresponding sides (Fig. 2). Then we developed a Fortran program called “Geomet Generator” allowing the reading of the Gmsh mesh file (*.msh) and the construction of control volumes.
Geomet Generator generates a ‘Geomet.g’ file containing the control volumes Data structures. This file will be then used by our main program called “porous drying simulator”. In previous works the code was validated and for further details reader can refer to (Rzig et al., 2017).

The grid dependency is checked as to obtain the suitable mesh size for which the drying time does not depend upon the mesh size or is independent of the number of nodes. The effect of grid dependency on drying time is shown in Table 2.

| Number of nodes | Drying time     |
|-----------------|-----------------|
| 2900            | 17h             |
| 8700            | 15h50min        |
| 14500           | 15h46min        |

As seen in table 2, there were almost no differences in the numerical solutions in high number meshed cases over 8700 nodes. Therefore, the computational domain with 8700 nodes was used for all numerical simulations.

4. RESULTS AND DISCUSSIONS

This section is devoted to the numerical results obtained from the compilation of numerical model which is applied to the drying of porous brick.

To numerically simulate the three-dimensional phenomenon of drying of the portion of the brick, we use free mesh generator Gmsh to achieve three-dimensional meshing (Fig. 3).

The operating conditions are taken the same of a typical industrial drying process which are listed in Table 3.

| T_{amb} (°C) | T_{ini} (°C) | S_{ini} (%) | C_{vamb} | P_{amb} (atm) | P_{ini} (atm) |
|--------------|--------------|-------------|----------|---------------|---------------|
| 100          | 20           | 50          | 0        | 1             | 1             |

From the time evolution of temperature, liquid saturation and gaseous pressure for four nodes aligned along the z axis (Figs. 4-6) where local coefficients of heat transfer (htx) and mass transfer (hmx) are employed, we can clearly observe the three conventional drying phases:
- The transient heating phase
- The constant drying rate phase
- The decreasing drying rate phase:
  - First decreasing drying rate period
  - Second decreasing drying rate period

However, from Fig. 4, it is noted that the isenthalpic phase is longer for the nodes away from the hot air inlet section. This is mainly due to the variation of transfer coefficients which have high values in the vicinity of the inlet (z = 0) whereas they decrease rapidly away from this zone. Far from the input section, where the coefficients are much lower, a very short isenthalpic phase is observed.

A preliminary analysis of the slab temperature, gas pressure and water saturation profiles (Figs. 4–6) shows that the variability of heat and mass transfer coefficients with z-direction has an appreciable effect on these state variables.
In order to better observe the effect of the variability of the coefficients, we have shown in Figs. 7-8 the three-dimensional distribution of the temperature, the liquid saturation and the pressure in water in the medium at different times.

Figure 7 shows the displacement of the drying front dividing the porous medium into two regions: one wet and where the temperature is near the wet temperature and the other dry where the temperature increases significantly. The drying front appears earlier in the vicinity of the inlet section. This is due to the intensity of the transfers in the vicinity of the input. This figure shows that the mass and thermal transfers are not two-dimensional.

After 8h of drying, Fig. 8 shows the non-symmetry of the saturation field along the y-axis under the effect of gravity. This behavior can be explained by the gravity effect since water migrates by gravity to the bottom of the brick portion(Fig. 8 (b)).

5. CONCLUSION

The present work constitutes a contribution to the understanding of the phenomena involved in heat and mass transfer mechanisms during the drying process in a porous medium. A three-dimensional heat and mass transfer model based on Whitaker's theory has been established. We simulated the case of drying by entrainment, a porous medium (brick), dried by an air flow.

For the discretization of the equations, we have adopted the method of control volumes based on finite elements. The effect of the heat and mass transfer coefficients has been studied and it has been shown that it is necessary to take into account the effect of the dynamic and mass thermal boundary layers on the heat and mass transfer coefficients. Consequently, we can conclude that not considering the variability of heat and mass transfer coefficients could cause significant errors.
NOMENCLATURE

$C_a$  specific heat of the air (kJ/kgK)

$C_p$  specific heat at constant pressure (kJ/kgK)

$C_{vw}$  specific heat of the vapor (kJ/kgK)

$C_{sw}$  specific heat of the water (kJ/kgK)

$D_{AB}$  diffusion coefficient (m²/s)

$D$  holes diameter (m)

$g$  gravitational acceleration (m/s²)

$h_{cm}$  convective mass transfer coefficient (m/s)

$h_{ct}$  convective heat transfer coefficient (W/m²°C)

$K$  intrinsic permeability (m²)

$L_c$  characteristic length of brick (m)

$M_a$  molar mass of air (kg/mol)

$M_v$  molar mass of vapour (kg/mol)

$m$  evaporation rate (kg/s)

$n_e$  outward normal vector

$P$  pressure (Pa)

$P_C$  capillary pressure (Pa)

$P_{v2}$  partial pressure of saturated vapour (Pa)

$R$  gas constant kJ/kmol.K

$r$  characteristic magnitude that represents the average radius of curvature of the menisci if the retention forces of the liquid are of capillary origin

$S$  liquid saturation (%)

$T$  temperature (K)

$t$  time (s)

$u_a$  air velocity (m/s)

Greek symbols

$\varepsilon$  porosity

$\varepsilon_l$  volume fraction of liquid phase

$\mu$  dynamic viscosity (kg/ms)

$\nu$  kinematic viscosity (m²/s)

Subscripts

$0$  initial condition

$a$  air

$e$  effective

$x$  local

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