Comprehensive Analysis of the Coated Component from a FORD Engine

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The article is devoted to a comprehensive analysis of the coating of a component from the Ford Focus 1.0 l EcoBoost engine. It is one of a pair of materially identical aluminum gears connected by a drive pulley and a drive. Aluminum gear components are surface treated with a mixture of metal powders using PVD technology, where the goal of the coating is to increase the hardness, abrasion resistance and heat of the base material. The chemical composition of the basic material of the geared aluminum drive wheel of the FORD engine proves that it is a component made by thermoforming and corresponds to the alloy EN AW 6061 (AlMg1SiCu) according to the standard ČSN EN 573 1-3. The microstructure of the base material component exhibits fine intermetallic phases evenly distributed throughout the cross-section of the base material, without the occurrence of larger inclusions and/or porosity. The cross-sectional microstructure in the area of the coating demonstrates that the component has a continuous uniform surface layer of the coating formed without a defect and does not have a diffuse bond with the base material. The measurement of the coating thickness of the component shows a continuous surface layer formed by powder metallography, which ranges from 70.1 µm to 143.9 µm. The surface EDS of the surface proves that it consists of deposited tungsten carbides produced by powder metallography and the bonding material is cobalt. Area EDS analysis of the surface coating identified morphologically two distinct areas of dark and light, where tungsten carbides and cobalt are based, and only the tungsten carbide and cobalt carbide contents change. The oxygen and carbon content also changes in the dark and light areas.

Keywords: FORD engine, component, coating, base material, scanning electron microscopy, EDS analysis, metallographic analysis, chemical analysis.

1 Introduction

Ford Motor Company is producing a new version of the low-emission Model 1.0L EcoBoost engine (Fig. 1), which is the first family petrol car in Europe with CO₂ emissions of 99 g/km [1]. The Focus 1.0L EcoBoost has a fuel consumption of 4.3 liters/100 km, which is equipped with a specially calibrated version of the 1.0L direct-injection petrol engine with a power output of 100 hp (74 hp), will also include Ford ECOncetic Technology including Auto-Start-Stop, Smart Regenerative Charging, Active Grille Shutter and Ford EcoMode, which support a more economical and environmentally friendly driving style. The Ford EcoBoost engine has a transverse three-cylinder direct-injection turbocharger with Ti-VCT. It uses a low-inertia turbocharger, split cooling system and direct fuel injection, which provide a surprising level of performance in the three-cylinder engine block.

The 1.0L Ford EcoBoost engine (Fig. 1) allows the new model to be more powerful than the first-generation Ford Focus with a 1.6L engine from less than 10 years ago and produce 47% less CO₂.

The component from the Ford Focus 1.0L EcoBoost engine was analyzed, namely one of a pair of materially identical aluminum gears connected by a drive pulley and a drive (Fig. 2 - marked with arrows). Aluminum gear components are surface treated with a mixture of metal powders using PVD technology [2-
The goal of the coating is to increase the hardness, abrasion resistance and heat of the base material. With PVD technology, there is no or only a very limited diffusion bonding layer between the coating and the substrate (coated material), a sharp interface remains and the adhesion forces are realized only by the physical principle.

2 Description of the analyzed component and analysis of the basic material

One gear connected to a drive pulley was used for analysis. Samples were taken from the coated wheel in the whole cross section (Fig. 3, 4) for chemical analysis of the base material. Layer thickness measurements were also performed as well as samples taken for EDS analysis of the coating composition on a scanning electron microscope.

The analysis of the basic material of the component was performed using a Q4 spark spectrometer at three different locations. The chemical composition of the base material of the aluminum gear wheel is shown in Tab. 1 as the average of three measurements. The chemical composition proves that it is a component made by thermoforming and corresponds to the alloy EN AW6061 (ISO AlMg1SiCu) according to the standard ČSN EN 5731-3.

In terms of chemical composition, it is an AlMgSi alloy with a low copper content as an accompanying (alloying) element [9-14]. The addition of a small amount of Cu will allow the strength of the base aluminum material to be increased by heat treatment by precipitation hardening.

3 Metallographic analysis of the part in section and measurement of the coating thickness

To examine the microstructure of the part, three metallographic sections were prepared, where one of them was taken to examine the structure of the parent metal and two other to measure the thickness of the coating [15]. Examination of the microstructure, coating compactness and coating thickness measurement was performed using an Olympus LEXT OLS 3100 confocal microscope. The microstructure of the component shows fine intermetallic phases of maximum occurrence of larger inclusions and/or porosity (Fig. 5). In terms of microstructure and gear shape, the part was made by profile extrusion in the shape of a profile. The microstructure in the cross-section in the area of the coating shows that the part has a continuous surface layer formed without a defect and even in the area of the tooth arc (Fig. 6 - right part of the coating) there is no reduction in the coating thickness.
Measurement of the coating thickness of the component at location A (Fig. 7) demonstrates in cross section that the component exhibits a continuous surface layer formed by powder metallography, which ranges in a narrow range from 70.1 µm to 95.6 µm (Tab. 2). The measurement was performed at 10 different locations and the measurement results are shown in the Table 2.

| Measurement number | 1  | 2  | 3  | 4  | 5  | 6  | 7  | 8  | 9  | 10 | Average value |
|--------------------|----|----|----|----|----|----|----|----|----|----|--------------|
| Measured layer thickness [µm] | 95.6 | 91.1 | 78.8 | 83.6 | 93.8 | 70.1 | 70.8 | 71.1 | 80.8 | 74.3 | 81.0 |

Measurement of the coating thickness of the component at location B (Fig. 8) shows in cross section that the component has a continuous surface layer formed by powder metallography, which ranges in a narrow range from 96.5 µm to 143.9 µm (Tab. 3). The measurement was performed in 10 different places and the measurement results are in the Tab. 3.

| Measurement number | 1  | 2  | 3  | 4  | 5  | 6  | 7  | 8  | 9  | 10 | Average value |
|--------------------|----|----|----|----|----|----|----|----|----|----|--------------|
| Measured layer thickness [µm] | 130.1 | 118.6 | 116.3 | 96.5 | 118.8 | 129.1 | 122.5 | 108.5 | 143.9 | 101.4 | 118.6 |

Analysis of the coating using scanning electron microscopy and EDS analysis

A Tescan VEGA 3 scanning electron microscope equipped with a Bruker EDX analyzer was used for chemical analysis of the coating and surface morphology. As part of the microscopic analysis by electron microscopy, the surface coating was analyzed from selected locations. In terms of morphology and different local coloration of the coating, two different sites can be identified, which were subsequently analyzed by EDX analysis. In terms of coating morphology, two distinct sites can be identified, namely irregular light areas that occupy about 60% of the coating surface.
and dark irregular areas that occupy about 40% of the component coating surface (Figs. 9, 10). Both different areas were then subjected to area EDX analyzes.

Area EDX analysis of the dark part of the surface proves that it is formed by powder-coated tungsten carbides and the bonding material is cobalt. From the point of view of quantitative analysis, it can be stated that there is more tungsten carbide in the dark area, where the analysis shows 35.33% of tungsten and 40.23% of carbon. The binding cobalt is 4.28% and there is also a high oxygen content is 16.79%, which is related to the coating technology and the oxide surface of the aluminum profile.

EDS coating analysis - black area

Tab 4 shows the elements that are identified from EDS analysis of the region visible on Fig. 12.
**Tab. 4** Results of EDX analysis from the area visible in Fig. 12

| Element | Series | unn. C [wt.%] | norm. C [wt.%] | Atom. C [at.%] | (3 Sigma) [wt.%] |
|---------|--------|---------------|----------------|----------------|-----------------|
| Al      | K-series | 0.46          | 0.42           | 0.32           | 0.17            |
| O       | K-series | 18.51         | 16.79          | 21.96          | 9.48            |
| W       | K-series | 38.94         | 35.33          | 4.02           | 3.03            |
| F       | K-series | 1.18          | 1.07           | 1.18           | 1.26            |
| Co      | K-series | 4.72          | 4.28           | 1.52           | 0.47            |
| Cr      | K-series | 1.68          | 1.53           | 0.61           | 0.24            |
| Na      | K-series | 0.38          | 0.35           | 0.32           | 0.20            |
| C       | K-series | 44.34         | 40-23          | 70-07          | 19.70           |
| Total:  |         | 110.21        | 100.00         | 100.00         |                 |

**EDS coating analysis - bright area**

![EDS analysis image](image_url)

Area EDX analysis of the light part of the surface proves that it consists of powder-coated tungsten carbides and the bonding material is cobalt. From the point of view of quantitative analysis, it can be stated that there is more tungsten carbide in the dark area, where the analysis shows 66.63 wt. % of tungsten and 16.54 wt. % of carbon. The binding cobalt is 6.39% by volume and there is also a high oxygen content of 7.89%, which is related to the coating technology and the oxide surface of the aluminum profile. The light area contains almost 100% more tungsten and less carbon, indicating a higher tungsten carbide content compared to the dark area. The binding area of cobalt also contains more bright area and contains about 100% less oxygen.

![EDS analysis image](image_url)

**Fig. 14** Surface EDX analysis from visible area in Fig. 13

Tab 4 shows the elements that are identified from EDS analysis of the region visible on Fig. 14.

**Tab. 5** Results of EDX analysis from the area visible in Fig. 14

| Element | Series | unn. C [wt.%] | norm. C [wt.%] | Atom. C [at.%] | (3 Sigma) [wt.%] |
|---------|--------|---------------|----------------|----------------|-----------------|
| Al      | K-series | 0.39          | 0.34           | 0.53           | 0.20            |
| O       | K-series | 8.99          | 7.89           | 20.53          | 7.87            |
| W       | K-series | 75.97         | 66.63          | 15.09          | 6.02            |
| C       | K-series | 18.86         | 16.54          | 57.33          | 14.62           |
| Co      | K-series | 7.29          | 6.39           | 4.51           | 0.78            |
| Cr      | K-series | 2.24          | 1.97           | 1.57           | 0.36            |
| Na      | K-series | 0.28          | 0.24           | 0.44           | 0.23            |
| Total:  |         | 114.02        | 100.00         | 100.00         |                 |
5 Conclusion

The chemical composition of the basic material of the geared aluminum drive wheel of the FORD engine proves that it is a component made by thermoforming and corresponds to the alloy EN AW 6061 (AlMg1SiCu) according to the standard ČSN EN 573 1-3. The microstructure of the component of the base material shows fine intermetallic phases up to a size of max. 10 - 15 mm. All the intermetallic phases are evenly distributed in the whole cross-section of the base material of the component, without the occurrence of larger inclusions and/or porosity.

In terms of microstructure and gear shape, the part was made by profile extrusion in the shape of a profile. The cross-sectional microstructure in the region of the coating shows that the component has a continuous uniform surface layer of the coating formed without a defect. Even in the region of the toothed arc and does not have a diffuse connection with the base material.

The measurement of the coating thickness of the component was performed at two different locations, and the results demonstrate that the component exhibits a continuous surface layer formed by powder metallurgy that ranges in the narrow range from 70.1 μm to 143.9 μm. The surface EDX of the surface proves that it consists of deposited tungsten carbides produced by powder metallurgy and the bonding material is cobalt. Area EDX analysis of the surface coating identified morphologically two distinct areas of dark and light, where tungsten carbides and cobalt are based, and only the tungsten carbide and cobalt carbide contents change. The oxygen and carbon content also changes in the dark and light areas.

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