Physiological Responses of a Model Marine Diatom to Fast pH Changes: Special Implications of Coastal Water Acidification

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Abstract

Diatoms and other phytoplankton in coastal waters experience rapid pH changes in milieu due to high biological activities and/or upwelled CO2-rich waters. While CO2 concentrating mechanisms (CCMs) are employed by all diatoms tested to counter low CO2 availability in seawater, little is known how this mechanism responds to fast pH changes. In the present study, the model diatom *Thalassiosira pseudonana* was acclimated for 20 generations to low pH (7.81) at an elevated CO2 of 1000 μatm (HC) or to high pH (8.18) at ambient CO2 levels of 390 μatm (LC), then its physiological characteristics were investigated as cells were shifted from HC to LC or vice versa. The maximal electron transport rate (ETRmax) in the HC-acclimated cells was immediately reduced by decreased CO2 availability, showing much lower values compared to that of the LC-acclimated cells. However, the cells showed a high capacity to regain their photochemical performance regardless of the growth CO2 levels, with their ETRmax values recovering to initial levels in about 100 min. This result indicates that this diatom might modulate its CCMs quickly to maintain a steady state supply of CO2, which is required for sustaining photosynthesis. In addition, active uptake of CO2 could play a fundamental role during the induction of CCMs under CO2 limitation, since the cells maintained high ETR even when both intracellular and periplasmic carbonic anhydrases were inhibited. It is concluded that efficient regulation of the CCM is one of the key strategies for diatoms to survive in fast changing pH environment, e.g. for the tested species, which is a dominant species in coastal waters where highly fluctuating pH is observed.

Introduction

In coastal waters, strong diel pH variations are often observed due to high biological production, upwellings or riverine input [1,2]. Superimposed on such natural pH variations, global oceans are being acidified due to continuously increasing atmospheric CO2 concentration and its subsequent increased dissolution, which will lead to an overall pH drop in the oceans of 0.3–0.4 units by 2100 [3,4]. In combination with eutrophication and hypoxia events, acidification may occur at a faster rate in coastal waters compared to open oceans [5]. Therefore, short-
term pH fluctuations may become larger with progressive ocean acidification. Such changes in the chemical environment may affect cytosolic pH [6] (and references therein) and cell homeostasis [7–9]. Therefore, it is of significance to examine the physiological responses of marine organisms to rapid pH changes in addition to their adaptive and evolutionary responses.

Marine diatoms contribute about 40% of oceanic primary productivity, playing a key role in carbon export to deep ocean waters and consequent regulation of global climate change [10,11]. In addition, diatoms also play an important role in the marine food web due to their high abundance and wide size distribution [12]. Therefore, how diatoms respond to ocean acidification (OA) or multiple environmental forcings is of general significance. However, documented findings on the effects of OA have been controversial, showing OA either caused enhanced growth [13,14], had neutral effects or inhibited growth of diatoms [15]. As one of the most sensitive mechanisms to changes in pCO₂, the CO₂ concentrating mechanism (CCM) has been suggested to be associated with a range of physiological processes [16]; consequently down regulation of CCMs under OA could have beneficial or detrimental effects under different light levels or treatment regimes [17,18].

In coastal waters and upwelling areas with highly fluctuating pH, diatoms are often the dominant representative phytoplankton group [2]. In these areas, pH fluctuations can exceed the 0.4 units predicted for the end of the century (and the same is true for corresponding CO₂ levels [2]. While they experience pH variations, the cells usually suffer from frequent CO₂ limitation, over short time scales, that can be a selective pressure for phytoplankton with active CCMs [9]. It is still unclear how diatoms may regulate their physiology to respond to rapid changes in pH and related changes in carbonate chemistry, and maintain a balance between photosynthetic efficiency and energy cost [19]. CCM related genes have been shown to respond to CO₂ changes within 1 hr when cyanobacteria were exposed to CO₂-free conditions [20]. However, since the downstream syntheses of CCM components should take longer if changes in gene expression are involved, how fast physiological responses to changes in carbonate chemistry occur remains unknown.

Since diatoms might activate anti-stress mechanisms quickly to cope with the fast chemical changes in coastal waters, to maintain CO₂ supply when cells encounter carbon shortage [21]. We therefore chose *Thalassiosira pseudonana*, a model diatom species frequently used for research in the past decades, to study whether diatoms can regulate physiological processes over a short timescale to respond to fast changes in pH/pCO₂. This species is distributed in coastal waters as well as open oceans. When grown under OA conditions, the growth rate shows little response under low light [22,23] but suffers from inhibition under high light levels [17,24]. Here, we show that this diatom can modulate its photosynthetic physiology to quickly respond to pH/pCO₂ changes, which may provide an advantage that allows it to outcompete other species in coastal waters.

**Materials and Methods**

**Species and culture condition**

*Thalassiosira pseudonana* (CCMP 1335), originally isolated from Moriches Bay, Long Island, USA, was inoculated in Aquil artificial seawater medium with a salinity of 35‰ [25], and grown in triplicate independent cultures (500 mL Erlenmeyer flasks) in a plant growth CO₂ chamber (HP1000G-D, Ruihua) at 20 ± 0.1°C with a 14 h:10 h light: dark cycle. The cultures were illuminated at 120 μmol photons m⁻² s⁻¹ provided by cool white fluorescent lamps, and partially renewed with CO₂ pre-equilibrated medium every day to maintain cell concentrations within a range of 8×10⁴ to 3×10⁵ cell mL⁻¹. Target pH (pCO₂) values in the cultures and the fresh medium were achieved by bubbling air (390 μatm) or pre-mixed air-CO₂ mixtures.
(1000 μatm) within the plant growth chamber, which controlled the high CO₂ level with a variation of less than 30 μatm. Cultures were acclimated under the respective CO₂ level for nine days (~20 generations).

**Determination of seawater carbonate chemistry**

pH was measured prior to and after the daily dilution as well as at the middle of the light period by a pH meter (pH510, Oaklon) which was calibrated with NBS buffer, to assure the stability of the carbonate system in cultures. Dissolved inorganic carbon (DIC) was sampled regularly and measured with an automatic system (AS-C3, Apollo SciTech Inc.) using an infrared gas analyzer (IRGA; Li-Cor7000, Li-Cor) after the samples were filtered into a syringe without any water/air CO₂ exchange. Samples of culture medium (0.5 mL) were acidified using phosphoric acid, and any subsequently released CO₂ was quantified by the IRGA. Parameters of the carbonate system were calculated as described in [13]. The carbonate chemical parameters were maintained at relatively stable levels, with pH changes between dilutions being less than 0.04 units (Table 1).

**Chlorophyll fluorescence measurements**

To assess the photochemical responses of the cells to changes in the seawater carbonate chemistry, 50 mL sub-culture of LC- or HC-acclimated cultures was aseptically taken from each culturing bottle during the middle of the light period, and placed in a water bath to control temperature at ~20°C. 5 mL culture (LC-acclimated or HC-acclimated) was then dispensed into 20 mM Tris buffered medium with different pH levels (pH_{NBS}, i.e. 7.82, 8.10, 8.37 and 9.50), and illuminated in the same growth chamber. Rapid light curves (RLCs) were then measured using a Xenon-Pulse Amplitude Modulated Fluorometer (XE-PAM, Walz). To assess the contribution of diffusive CO₂ entry to photosynthetic efficiency, sub-samples were pipetted directly from illuminated cultures during the middle of light period, dispensed into 20 mM Tris buffered seawater medium that was pre-adjusted to target pH with or without ethoxyzolamide (EZ), a membrane permeable inhibitor of carbonic anhydrase, at a final concentration of 200 μmol L⁻¹ (pre-tests showed that 150 μmol L⁻¹ is a sufficient concentration).

The RLC was determined as relative electron transport rate (rETR) in response to eight different and increasing actinic PAR intensities, with a 10 s duration for each increment separated by a 0.8 s saturating pulse (4000 μmol m⁻² s⁻¹).

**Determination of activity of carbonic anhydrase**

Cells grown under the two pCO₂ levels were harvested by gentle filtration onto a polycarbonate membrane, washed and re-suspended (cell concentration around 4×10⁶ mL⁻¹) in DIC free seawater medium. The major parameters of seawater carbonate system under the present and projected CO₂ levels are shown in Table 1. Data are the means ± SD of 6 measurements from triplicate bottles before and after dilution in a typical day.

| parameter      | LC-acclimated | HC-acclimated |
|----------------|---------------|---------------|
| pH             | 8.18±0.04     | 7.81±0.03     |
| DIC (μmol kg⁻¹) | 2023±28       | 2184±21       |
| CO₂ (μmol kg⁻¹) | 12.6±1.4      | 33.6±2.4      |
| HCO₃⁻ (μmol kg⁻¹) | 1828±19     | 2063±17       |
| CO₃²⁻ (μmol kg⁻¹) | 182±17      | 87.7±9        |
| pCO₂ (µatm)    | 390±41        | 1033±72       |

Table 1. The major parameters of seawater carbonate system under the present and projected CO₂ levels. Data are the means ± SD of 6 measurements from triplicate bottles before and after dilution in a typical day.

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seawater that was buffered with 20 mM barbitone at pH 8.4. The CAe (external CA) and CAtotal (including periplasmic and intracellular CA) activity of the intact cells was determined by an electrometric method according to [26]. A 5 mL cell suspension (for CAext) or cell suspension disrupted with a sonicator (102C Branson, for CAtotal) with amplitude set at 30%, then added to a reaction cuvette and stirred. The disruption of cells was confirmed under a microscope. The time required for the pH to drop from 8.2 to 7.2 after the addition of 2 mL ice-cold CO2-saturated pure water was recorded. The temperature during the reaction was controlled at 4°C.

Calculations and statistical analysis

The rETR was calculated as: 

\[
 rETR = \Phi_{PSII} \times 0.5 \times PFD \ (\mu\text{mol e}^{-}\text{m}^{-2}\text{s}^{-1}),
\]

where \( \Phi_{PSII} \) is the photochemical quantum yield of PSII in light, PFD is the actinic PAR intensity (\( \mu\text{mol photons m}^{-2}\text{s}^{-1} \)), and the factor 0.5 accounts for approximately 50% of all the absorbed energy being allocated to PSII. Rapid light curves were fitted according to [27] as follows:

\[
y = \frac{x}{(ax^2 + bx + c)},
\]

Where y is the rETR, x is the photon flux density of actinic light (\( \mu\text{mol photons m}^{-2}\text{s}^{-1} \)), a, b and c are the adjustment parameters. \( P_{m}, \alpha \) and \( E_k \) was calculated as:

\[
 P_{m} = \frac{1}{(b + 2\sqrt{ac})}, \quad \alpha = \frac{1}{c}, \quad E_k = P_{m}/a
\]

Activity of the carbonic anhydrase (in enzyme units, EU) was calculated as follows:

\[
10 \times \left\{ \frac{(T_b/T_c)-1}{10^6 \text{ cells}} \right\}
\]

where \( T_b \) and \( T_c \) were the times in seconds for the pH drop with or without cells, respectively.

Statistical differences among treatments were tested with one-way ANOVA using a Tukey test by Origin 8.0, and the significance level was set at \( p = 0.05 \).

Results

The rapid light curves (RLCs) showed typical patterns of photosynthesis versus irradiance; relative electron transport rate (rETR) increased with increasing light (1), while significant inhibition of rETR by EZ was observed for RLCs measured at both pH 7.82 and pH 8.37 (Fig 1). rETR values of LC-acclimated and HC-acclimated cells treated with EZ (LC-acclimated +EZ or HC-acclimated +EZ cells) were significantly lower than those of the controls (without EZ) \( p<0.001 \), and recovered partially after 22–100 min incubation. The rETR of LC-acclimated cells was similar to that of HC-acclimated cells when measured at pH 7.82 (Fig 1A–1D), while when samples were assayed at pH 8.37, the rETR of HC-acclimated cells were lower than that of LC-acclimated cells in the first 22 min (Fig 1E and 1F), but achieved similar values after 44 min incubation (Fig 1G and 1I). After 100 min incubation, rETR values of EZ treated samples was still lower than the control treatments when assayed at pH 7.82, while no significant differences were found among samples that were assayed at pH 8.37. Similar responses were observed at pH 8.10; rETR rates were significantly inhibited by EZ but recovered partially during the incubation (Fig A–D in S1 Fig) \( p<0.001 \). However the responses were completely different at pH 9.50 where essentially all the DIC was available only as HCO\(_3\)-. Here no differences were found among treatments, and no further recovery occurred during the incubation (Fig E–I in S1 Fig).

The maximal relative electron transport rate (rETR\(_m\)) measured at pH 7.82 remained relatively stable for LC-acclimated and HC-acclimated cells, values for both being around 128 plusmn;4 \( \mu\text{mol e}^{-}\text{m}^{-2}\text{s}^{-1} \) during the 100 min incubation (Fig 2A). The rETR\(_m\) of LC-acclimated +EZ cells showed a similar pattern, with relatively lower values around 113±2 \( \mu\text{mol e}^{-}\text{m}^{-2}\text{s}^{-1} \). While the rETR\(_m\) of HC-acclimated +EZ cells was around 73% of that of LC-acclimated +EZ cells for the first 22 min, then recovered quickly to similar values as LC-acclimated + EZ cells for the subsequent 78 min of incubation (Fig 2A). There were obvious differences between LC-acclimated and HC-acclimated cells at pH 8.37 when CO\(_2\) availability was only
Fig 1. The rapid light curves of LC-acclimated and HC-acclimated grown cells in the presence or absence of EZ measured in pH 7.82 (A, B, C, D) or pH 8.37 (E, F, G, I) buffered medium (Tris, 20 mM) at different times (T₀, T₂₂, T₄₄ and T₁₀₀). Vertical bars represent SD, n = 3.

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25% of that at pH 7.82 (Fig 2B). \( \text{rETR}_m \) values of LC-acclimated cells measured at pH 8.37 decreased slightly but significantly during the 100 min incubation \((p<0.01)\), while HC-acclimated cells had \( \text{rETR}_m \) values significantly lower (by 20%) than those of LC-acclimated cells at \( T_0 \) \((p<1\times10^{-4})\), but increased linearly to reach similar values to those of LC-acclimated cells by the end of the incubation. The \( \text{rETR}_m \) values of HC-acclimated +EZ cells were 19% lower than those of LC-acclimated +EZ cells, but during the incubation, all increased gradually to final values that were close to those of non EZ treated samples (Fig 2B).
The $E_k$ values of LC-acclimated and HC-acclimated cells at pH 7.82 remained relatively stable during the incubation, at $780\pm 43 \mu \text{mol m}^{-2} \text{s}^{-1}$. The $E_k$ of LC-acclimated +EZ cells was $619\pm 26 \mu \text{mol m}^{-2} \text{s}^{-1}$ initially but increased slightly to $672\pm 24 \mu \text{mol m}^{-2} \text{s}^{-1}$ then remained stable for the subsequent incubation (Fig 2C) ($p = 0.058$). The $E_k$ of HC-acclimated +EZ cells was the lowest at $T_0$, being around 77% of that of LC-acclimated +EZ cells, but increased gradually and reached a similar value to that found in LC-acclimated +EZ cells at the end of incubation (Fig 2C). The $E_k$ measured at pH 8.37 remained relatively stable for the whole incubation at around $500\text{–}650 \mu \text{mol m}^{-2} \text{s}^{-1}$ with some variations but no clear trend (Fig 2D). The values of light utilization efficiency ($\alpha$, initial slope of RLC) of all treatments were around $0.18 \text{ e}^{- \text{photon}^{-1}}$ at $T_0$ under pH 7.82, and remained stable in most treatments for the subsequent incubation (Fig 2E). While values for HC-acclimated +EZ cells decreased significantly from $0.17 \pm 0.009$ to $0.15\pm 0.005 \text{ e}^{- \text{photon}^{-1}}$ at $T_22$ ($p < 0.05$), and then increased gradually to a similar value as the other treatments (Fig 2E). $\alpha$ of LC-acclimated and HC-acclimated cells at pH 8.37 was similar as to that at pH7.82, and remained stable for the whole incubation period, at around 0.17, though $\alpha$ of LC-acclimated +EZ and HC-acclimated +EZ cells was lower at $T_0$, at $0.15\pm 0.007$ and $0.11\pm 0.008 \text{ e}^{- \text{photon}^{-1}}$, respectively (Fig 2F). While EZ treated cells all gradually increased to a similar value to that found in non-EZ samples at $T_{100}$, RLC parameters measured at pH 8.10 and pH 9.50 are shown in S2 Fig.

The $rETR_m$ values across the assay pH showed a decreasing pattern with increasing pH levels, at $T_0$ (Fig 3A). $rETR_m$ values of LC-acclimated cells was similar to those of HC-acclimated cells at pH 7.82. While HC-acclimated values were lower than those of LC-acclimated at higher pH, around 80% of LC-acclimated at pH8.37, but all decreased to the same value, around $60 \mu \text{mol e}^{- \text{m}^{-2} \text{s}^{-1}}$, when measured at pH 9.50. $rETR_m$ of HC-acclimated EZ cells was lower than LC-acclimated EZ, except for at pH 9.50 where both decreased to similar values to those of LC-acclimated or HC-acclimated cells. At $T_{100}$, $rETR_m$ of LC-acclimated and HC-acclimated cells were similar to those at $T_0$, and dropped from $125\pm 2$ to $38\pm 2 \mu \text{mol e}^{- \text{m}^{-2} \text{s}^{-1}}$ when the assay pH gradually increased from 7.82 to 9.50. $rETR_m$ values of EZ treated cells were slightly lower than for non EZ samples, and decreased with increasing pH to a value of $38\pm 2 \mu \text{mol e}^{- \text{m}^{-2} \text{s}^{-1}}$ at pH 9.50. The light utilization efficiency values of LC-acclimated and HC-acclimated cells were similar and stable, around $0.18 \text{ e}^{- \text{photon}^{-1}}$, while the EZ treated samples had lower values at pH8.10 and pH8.37, but were similar to non-EZ samples at pH7.82 or pH9.50 (Fig 3E and 3F). The light utilization efficiency values at $T_{100}$ were similar and stable for all treatments at pH 7.82 to pH 8.37, while they decreased sharply by 50% at pH 9.50. No differences were found among CO2 or EZ treatments at $T_{100}$.

The whole cell carbonic anhydrase activity ($CA_{\text{total}}$) was $0.18\pm 0.03 \text{ EU (10}^6 \text{ cells})^{-1}$ for LC-acclimated cells, while that of HC-acclimated cells was much less ($p<0.01$), around $0.09\pm 0.05 \text{ EU (10}^6 \text{ cells})^{-1}$ (Fig 4B). Extracellular carbonic anhydrase ($CA_e$) activity was detectable but very low for both LC-acclimated and HC-acclimated cells, at $0.02 (\pm 0.01)$ and $0.01 (\pm 0.004) \text{ EU (10}^6 \text{ cells})^{-1}$ respectively.

**Discussion**

Fluctuations of environmental factors are regarded as providing selection pressure for dominant species of phytoplankton [28]. As the oceans continue to absorb anthropogenic CO2, the buffering capacity of seawater will decrease, therefore phytoplankton cells are likely to experience increasing acidic stress with progressive ocean acidification [5,29]. This will be superimposed on the rapid and significant fluctuations in pH and CO2 concentration experienced by phytoplankton in upwelling regions and near-shore or estuarine environments[2]. In the present study, by comparing the physiological responses of a diatom, grown at two $pCO_2$ levels, to
rapid changes of pH, we concluded that *T. pseudonana* could respond quickly to changes of pH, to maintain a steady state supply of CO2. When LC-acclimated cells are exposed to high pH (limited CO2) environments, they could immediately achieve a relatively high electron transport rate, while there was an obvious lag before HC-acclimated cells responded to high pH. This may provide this diatom with a competitive advantage in coastal waters [30].

Short-term exposure to low pH/high CO2 should lead to negligible enhancement of photosynthesis of the LC-acclimated cells due to a sufficient supply of CO2 by the CCM [31], recognizing that in *T. pseudonana* a biochemical CCM based on C4 photosynthesis [32–36], rather...
than the biophysical CCM found in most other microalgae [31] may operate. The diatom tested in this study, when acclimated to LC conditions, can achieve higher electron flow than HC-acclimated cells [13], which could be attributed to the higher uptake of CO₂ for the photosynthetic machinery [37]. Once the whole cellular enzymatic conversion between CO₂ and HCO₃⁻ was blocked by the inhibitor EZ, the rETR sharply decreased due to an inhibited CO₂ supply; in such a case, the photosynthetic rate should solely depend on diffusive entry or active uptake of CO₂. The higher rETR of LC-acclimated +EZ cells compared to that of HC-acclimated +EZ cells, suggests that when CO₂ was the only source of inorganic carbon, LC-acclimated cells had a higher efficiency in active CO₂ uptake. In T. pseudonana, the C4 pathway could play an important role during the acclimation under low CO₂ or the diel cycle [32], since previous studies have demonstrated that T. pseudonana has the ability to concentrate CO₂ via the C4 (or a C3-C4 intermediate) pathway for photosynthesis [33,34]. Indeed, our results are in agreement with a recent study, which revealed that C4 pathway-related genes were up-regulated during an acclimation under reduced CO₂ availability [35], indicating a fundamental role for the C4 pathway in T. pseudonana photosynthesis, especially in a low CO₂ environment [36].

The diminishing differences of rETRₘ with time among the treatments with or without EZ in HC-acclimated or LC-acclimated cells (Fig 2A and 2B), indicated that there was an inducible mechanism operational for the carbon uptake once the cells encountered CO₂ shortage [34,36], especially for EZ treated cells, and that this allowed recovery of photosystem II. Since light energy is essential for the operation of the CCM [38,39], when exposed to high pH with limited supply of CO₂, the cells have to allocate extra energy for carbon transport or synthesis of C4 compounds, resulting in a lower light utilization efficiency (α) (Fig 2F). Even when major components of CCMs, extra- and intracellular carbonic anhydrase, were blocked with sufficient inhibitor, rETRₘ values of EZ cells could recover by ~40% to ~90% of non-EZ treated cells.
samples within 100 min, indicating that the cells could modulate their physiological processes to adapt to extreme conditions [36], including e.g. the up-regulation of the C4 pathway, reallocation of light energy to CCM, active CO$_2$ uptake, to maintain steady state CO$_2$ supply [19].

Phytoplankton experience fluctuating environmental changes due to their temporal-spatial distributions. Riverine input, mixing or cyclones could stimulate the growth of some species which have strong nutrient accumulating mechanisms [40,41], while fluctuating sunlight would influence photosynthetic carbon fixation in surface seawater depending on levels of solar radiation [42,43] and phytoplankton cells circulating in the water column will be exposed to rapidly changing light levels. Therefore, phytoplankton species may have acquired, during evolution, sophisticated mechanisms to modulate their physiology to respond to fast environmental changes. As revealed recently, the chain forming diatom could maintain photochemical performance, even exposed to acute pH changes [44]. CO$_2$ is the major factor regulating CCM activity; while some diatoms rely on CO$_2$ diffusion more than active transport of bicarbonate across the plasma membrane [45], the varied CO$_2$ availability in coastal waters may affect their photosynthesis. Based on the findings of this work, we see that diatoms like *T. pseudonana* can cope with frequent fluctuations of pH/CO$_2$, and maintain CO$_2$ supplies at a steady state for photosynthesis. This can be considered as an advantage for these species, allowing them to dominate coastal waters [46].

**Supporting Information**

**S1 Fig.** The rapid light curves of LC-acclimated and HC-acclimated grown cells in the presence or absence of EZ that were measured in pH 8.10 (Fig A, B, C, D) or in pH 9.50 (Fig E, F, G, I) buffered medium (Tris, 20 mM) and at different time (T$_0$, T$_{22}$, T$_{44}$ and T$_{100}$min). Vertical bars represent SD, n = 3. (TIF)

**S2 Fig.** The time course of photosynthetic parameters (P$_m$, the maximal rETR; E$_{k}$, the light saturation point and $\alpha$, the light utilization efficiency) of LC-acclimated and HC-acclimated rapid light curves in the presence or absence of EZ at pH 8.10 (Fig A, C, E) or pH 9.50 (Fig B, D, F), Vertical bars represent SD, n = 3. (TIF)

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**Author Contributions**

Conceived and designed the experiments: YW KG. Performed the experiments: YW. Analyzed the data: YW. Contributed reagents/materials/analysis tools: YW. Wrote the paper: YW JB KG.
References

1. Cornwall CE, Hepburn CD, McGraw CM, Currie KL, Pilditch CA, Hunter KA, et al. Diurnal fluctuations in seawater pH influence the response of a calcifying macroalga to ocean acidification. Proc R Soc B Biol Sci. 2013; 280: 20132201. doi: 10.1098/rspb.2013.2201

2. Hofmann GE, Smith JE, Johnson KS, Send U, Levin LA, Micheli F, et al. High-frequency dynamics of ocean pH: A multi-ecosystem comparison. PLoS ONE. 2011; 6: e28983. doi: 10.1371/journal.pone.0028983 PMID: 22025986

3. IGBP, IOC, SCOR. 2013. Ocean acidification summary for policymakers-Third symposium on the ocean in a high-CO2 world. International Geosphere-Biosphere Programme.

4. Doney SC. The dangers of ocean acidification. Sci Am. 2006; 294: 58–65. PMID: 16502612

5. Cai W-J, Hu X, Huang W-J, Murrell MC, Lehrter JC, Lohrenz SE, et al. Acidification of subsurface coastal waters enhanced by eutrophication. Nat Geosci. 2011; 4: 766–770. doi: 10.1038/ngeo1297

6. Raven J. A. 2013. Half a century of pursuing the pervasive proton. In Progress in Botany, pp. 3–34. Ed. by Lüttge U., Bayschlag W., Francis D., and Cushman J. Springer Berlin Heidelberg.

7. Flynn K, Blackford JC, Baird ME, Raven JA, Clark DR, Beardall J, et al. Changes in pH at the exterior surface of plankton with ocean acidification. Nat Clim Change. 2012; 2: 510–513. doi: 10.1038/nclimate1489

8. Giordano M. Homeostasis: An underestimated focal point of ecology and evolution. Plant Sci. 2013; 211: 92–101. doi: 10.1016/j.plantsci.2013.07.008 PMID: 23987815

9. Fanesi A, Raven JA, Giordano M. 2014. Growth rate affects the responses of the green alga Tetraselmis suecica to external perturbations. Plant, Cell & Environment 37, 512–519.

10. Ducklow HW, Steinberg DK, Buesseler KO. 2001. Upper ocean carbon export and the biological pump. Oceanography 14, 50–58.

11. Scott KN. 2005. Day after Tomorrow: Ocean CO2 sequestration and the future of climate change, The. Georgetown International Environmental Law Review 18, 57.

12. Finkel ZV, Beardall J, Flynn KJ, Quigg A, Rees TAV, Raven JA. Phytoplankton in a changing world: cell size and elemental stoichiometry. J Plankton Res. 2010; 32: 119–137. doi: 10.1093/plankt/fbp098

13. Wu Y., Gao K. & Riebesell U. 2010. CO2-induced seawater acidification affects physiological performance of the marine diatom Phaeodactylum tricornutum. Biogeosciences 7, 2915–2923.

14. Wu Y, Campbell DA, Irwin AJ, Suggett DJ, Finkel ZV. Ocean acidification enhances the growth rate of larger diatoms. Limnol Oceanogr. 2014; 59: 1027–1034.

15. Gao K, Campbell DA. 2014. Photophysiological responses of marine diatoms to elevated CO2 and decreased pH: a review. Functional Plant Biology, 449–459.

16. Raven JA, Giordano M, Beardall J, Maberly SC. 2011. Algal and aquatic plant carbon concentrating mechanisms in relation to environmental change. Photosynthesis Research 109, 281–296. doi: 10.1007/s11120-011-9632-6 PMID: 21327536

17. Gao K, Xu J, Gao G, Li Y, Hutchins D, Huang B, et al. 2012 Rising CO2 and increased light exposure synergistically reduce marine primary productivity. Nat. Clim. Change 2, 519–523.

18. Hoppe C, Beszteri S, Bachmann J, Frickenhaus S, Holtz L-M, Trimborn S, et al. 2014. Dynamic light alters the responses of Chaetoceros debilis to ocean acidification. Ocean Sciences Meetings, Honolulu, Hawaii, USA, 2014-02-23-2014-02-28. Honolulu, Hawaii, USA.

19. Raven JA, Beardall J, Giordano M. Energy costs of carbon dioxide concentrating mechanisms in aquatic organisms. Photosynth Res. 2014; 1–14. doi: 10.1007/s11120-013-9962-7 PMID: 24390639

20. Woodger FB, Badger MR, Price GD. Inorganic carbon limitation induces transcripts encoding components of the CO2-concentrating mechanism in Synechococcus sp. PCC7942 through a redox-independent pathway. Plant Physiol. 2003; 133: 2069–2080. doi: 10.1104/pp.103.029728 PMID: 14645730

21. Hobson LA, Hanson CE, Holeton C. 2001. An ecological basis for extracellular carbonic anhydrase in marine unicellular algae. Journal of Phycology 37, 717–723.

22. Crawford KJ, Raven JA, Wheeler GL, Baxter EJ, Joint I. 2011. The response of Thalassiosira pseudonana to long-term exposure to increased CO2 and decreased pH. PLoS ONE 6, e26695. doi: 10.1371/journal.pone.0026695 PMID: 22053201

23. Yang G, Gao K. 2012. Physiological responses of the marine diatom Thalassiosira pseudonana to increased pCO2 and seawater acidity. Marine Environmental Research 79, 142–151. doi: 10.1016/j.marenvres.2012.06.002 PMID: 22770534

24. McCarthy A, Rogers SP, Duffy SJ, Campbell DA. 2012. Elevated carbon dioxide differentially alters the photophysiology of Thalassiosira pseudonana (bacillariophyceae) and Emiliania huxleyi (haptophyta). Journal of Phycology 48, 635–646.
25. Morel FMM, Rueter JG, Anderson DM, Guillard RRL. Aquil: A chemically defined phytoplankton culture medium for trace metal studies. J Phycol. 1979; 15: 135–141. doi:10.1111/j.1529-8817.1979.tb02976.x

26. Wilbur KM, Anderson NG. Electrometric and colorimetric determination of carbonic anhydrase. J Biol Chem. 1948; 176: 147–154. PMID: 1886152

27. Eilers PHC, Peeters JCH. A model for the relationship between light intensity and the rate of photosynthesis in phytoplankton. Ecol Model. 1988; 42: 199–215. doi:10.1016/0304-3800(88)90057-9

28. Descamps-Julien B, Gonzalez A. Stable coexistence in a fluctuating environment: an experimental demonstration. Ecology. 2005; 86: 2815–2824. doi:10.1890/04-1700

29. Beer S, Björk M, Beardall J. Photosynthesis in the marine environment. John Wiley & Sons; 2014.

30. Falkowski PG, Oliver MJ. Mix and match: how climate selects phytoplankton. Nat Rev Microbiol. 2007; 5: 813–819. doi:10.1038/nrmicro1751 PMID: 17853908

31. Kaplan A, Reinhold L. 1999. CO2 concentrating mechanisms in photosynthetic microorganisms. Annual Review of Plant Physiology and Plant Molecular Biology 50, 539–570. PMID: 15012219

32. Granum E, Roberts K, Raven JA, Leegood RC. Primary carbon and nitrogen metabolic gene expression in the diatom Thalassiosira pseudonana (bacillariophyceae): Diel periodicity and effects of inorganic carbon and nitrogen. J Phycol. 2009; 45: 1083–1092. doi:10.1111/j.1529-8817.2009.00728.x

33. Beardall J, Mukerji D, Glover HE, Morris I. The path of carbon in photosynthesis by marine phytoplankton. J Phycol. 1976; 12: 409–417. doi:10.1111/j.1529-8817.1976.tb02864.x

34. Reinfelder JR, Kraepiel AML, Morel FMM. Unicellular C4 photosynthesis in a marine diatom. Nature. 2000; 407: 996–999. doi:10.1038/35039612 PMID: 11069177

35. Kustka AB, Milligan AJ, Zheng H, New AM, Gates C, Bidle KD, et al. Low CO2 results in a rearrangement of carbon metabolism to support C4 photosynthetic carbon assimilation in Thalassiosira pseudonana. New Phytol. 2014; 204: 507–520. doi:10.1111/nph.12926 PMID: 25046577

36. Reinfelder JR, Milligan AJ, Morel FMM. The role of the C4 pathway in carbon accumulation and fixation in a marine diatom. Plant Physiol. 2004; 135: 2106–2111. doi:10.1104/pp.104.041319 PMID: 15286292

37. Sukenik A, Bennett J, Falkowski P. 1987. Light-saturated photosynthesis—limitation by electron transport or carbon fixation? Biochimica et Biophysica Acta 891, 205–215.

38. Fukuzawa H, Miura K, Ishizaki K, Kucho K, Saito T, Kohinata T, et al. 2001. CCM1, a regulatory gene controlling the induction of a carbon-concentrating mechanism in Chlamydomonas reinhardtii by sensing CO2 availability. Proceedings of the National Academy of Sciences 98, 5347–5352.

39. Yamano T, Tsujikawa T, Hatano K, Ozawa S, Takahashi Y, Fukuzawa H. 2010. Light and low-CO2-dependent LCIB–LCIC complex localization in the chloroplast supports the carbon-concentrating mechanism in Chlamydomonas reinhardtii. Plant and Cell Physiology 51, 1453–1468. doi:10.1093/pcp/pcq105 PMID: 20660228

40. Holligan PM, Pingree RD, Mardell GT. Oceanic solitons, nutrient pulses and phytoplankton growth. Nature. 1985; 314: 348–350. doi: 10.1038/314348a0

41. Lin I, Liu WT, Wu C-C, Wong GTF, Hu C, Chen Z, et al. New evidence for enhanced ocean primary production triggered by tropical cyclone. Geophys Res Lett. 2003; 30: 1718. doi: 10.1029/2003GL017141

42. Gao K, Wu Y, Li G, Wu H, Villafañe VE, Helbling EW. 2007. Solar UV radiation drives CO2 fixation in marine phytoplankton: a double-edged sword. Plant Physiology 144, 54–59. PMID: 17494919

43. Li G, Gao K. Differential impacts of solar UV radiation on photosynthetic carbon fixation from the coastal to offshore surface waters in the South China Sea. Photochem Photobiol. 2011; 87: 329–334. doi: 10.1111/j.1751-1097.2010.00862.x PMID: 21155829

44. Zheng Y., Giordano M., and Gao K. 2015. Photochemical responses of the diatom Skeletonema costatum grown under elevated CO2 concentrations to short-term changes in pH. Aquatic Biology, 23: 109–118.

45. Roberts K, Granum E, Leegood RC, Raven JA. Carbon acquisition by diatoms. Photosynth Res. 2007; 93: 79–88. doi: 10.1007/s11120-007-9172-2 PMID: 17497225

46. Price NM, Thompson PA, Harrison PJ. 1987. Selenium: An essential element for growth of the coastal marine diatom Thalassiosira pseudonana (bacillariophyceae). Journal of Phycology 23, 1–9.