Effects of the Heterointerface on the Growth Characteristics of a Brownmillerite SrFeO$_{2.5}$ Thin Film Grown on SrRuO$_3$ and SrTiO$_3$ Perovskites

Janghyun Jo$^1$, Venkata Raveendra Nallagatlla$^2$, Susant Kumar Acharya$^2$, Youngho Kang$^3$, Yoonkoo Kim$^1$, Sangmoon Yoon$^1$, Sangmin Lee$^1$, Hionsuck Baik$^1$, Seungwu Han$^1$, Miyoung Kim$^1$* & Chang Uk Jung$^2$*

Manipulation of the heterointerfacial structure and/or chemistry of transition metal oxides is of great interest for the development of novel properties. However, few studies have focused on heterointerfacial effects on the growth characteristics of oxide thin films, although such interfacial engineering is crucial to determine the growth dynamics and physical properties of oxide heterostructures. Herein, we show that heterointerfacial effects play key roles in determining the growth process of oxide thin films by overcoming the simple epitaxial strain energy. Brownmillerite (SrFeO$_{2.5}$; BM-SFO) thin films are epitaxially grown along the $b$-axis on both SrTiO$_3$(001) and SrRuO$_3$/SrTiO$_3$(001) substrates, whereas growth along the $a$-axis is expected from conventional epitaxial strain effects originating from lattice mismatch with the substrates. Scanning transmission electron microscopy measurements and first principles calculations reveal that these peculiar growth characteristics of BM-SFO thin films originate from the heterointerfacial effects governed by their distinct interfacial structures. These include octahedral connectivity between dissimilar oxides containing different chemical species and a peculiar transition layer for BM-SFO/SrRuO$_3$/SrTiO$_3$(001) and BM-SFO/SrTiO$_3$(001) heterostructures, respectively. These effects enable subtle control of the growth process of oxide thin films and could facilitate the fabrication of novel functional devices.

The diverse properties and broad utility of transition metal oxides having the ABO$_3$ perovskite structure and related structures result from strong coupling among the lattice, charge, spin, and orbital degrees of freedom. These basically originate from the strong hybridization between transition metal $d$ and oxygen 2$p$ orbitals$^{1,2}$. Engineering such complex interactions has been recognized as a way to tailor the functional properties of transition metal oxides. In recent years, manipulating the heterointerface of transition metal oxides has enabled the development of novel phenomena such as two-dimensional free electron gases$^{3-7}$, interfacial charge transfer$^{8,9}$, high-$T_c$ superconductors$^{10}$, and colossal magnetoresistance$^{11}$; these have never been available in bulk equilibrium phases.

These new phenomena emerged through controlling the interface, which essentially modify the degree of orbital hybridization$^{12-14}$. Such control is mediated by, for example, structural distortions, crystal symmetry, and oxygen coordination environments of oxide-based heterointerfaces. The interfacial structure and/or chemistry can also profoundly influence the growth dynamics of oxide thin films on foreign substrates. The resulting growth behaviours of oxide thin films would significantly affect the microstructures and physical properties of the thin films. Therefore, accurate determination of the oxide heterointerfacial structure is essential to understand the growth characteristics of oxide thin films and fabricate functional devices with desired physical properties.

$^1$Department of Materials Science and Engineering and Research Institute of Advanced Materials, Seoul National University, Seoul, 08826, Republic of Korea. $^2$Department of Physics and Oxide Research Centre, Hankuk University of Foreign Studies, Yongin, 17035, Republic of Korea. $^3$Seoul Center, Korea Basic Science Institute, Seoul, 136-713, Republic of Korea. *email: mkim@snu.ac.kr; cu-jung@hufs.ac.kr
Brownmillerite oxides such as SrFeO$_2.5$ (BM-SFO) are of particular interest due to their wide range of physical properties including thermostability, fast oxygen-ion transport, catalysis, and topolectric phase transformation at low temperatures$^{15,16}$. These properties allow BM oxides to be exploited for extensive applications such as in solid-oxide fuel cells, sensors, membranes for oxygen separation, photon catalysis, and memristors$^{17-19}$. BM oxides have a perovskite-derived structure with parallel rows of ordered oxygen vacancies that create tetrahedral chains directed along [101]$_{pc}$ in every second (010)$_{pc}$ octahedral plane. This results in three-dimensional A$_2$BO$_3$ structures composed of alternate stacking of BO$_6$ octahedral layers and sheets of BO$_4$ tetrahedral layers$^{20}$. Herein, the subscript ‘pc’ represents the perovskite unit cell in pseudocubic notation, which is derived from the intrinsic crystal structure of the BM phase$^{20}$.

Controlling the direction of the oxygen vacancy channels in the BM structure, i.e., the growth orientation of BM oxides, has recently become important in a wide range of scientific and industrial areas. The oxygen vacancy channels are believed to form a pathway for easy and rapid oxygen-ion transport and to contribute to the ionic conductivity of the compound in applications such as solid-oxide fuel cells and resistive-switching memory devices$^{20-22}$. The highly anisotropic nature of these layers originating from the ordered oxygen vacancy channels of the BM structure could alter the growth behaviour$^{21,22}$. Many studies have explored epitaxial strain to manipulate the growth orientation of BM thin films. They have demonstrated that oxygen-deficient channels of BM oxides can be arranged parallel or perpendicular to the substrate, depending on the interfacial strain resulting from lattice mismatch with the substrate$^{20,24}$. In addition to controlling the growth process of the thin films by imposing epitaxial strain alone, various interfacial effects can extend the range of choice for engineering microstructures as well as the physical properties. However, these effects have been rarely reported, despite their great potential for manipulating the growth dynamics and consequential physical properties of oxide thin films.

Herein, we report novel interfacial effects that have a significant influence on the growth behaviour of oxide thin films by overcoming the conventional epitaxial strain effects. We investigated the microstructures of BM-SFO thin films grown on SrTiO$_3$(STO)(001) and SrRuO$_3$(SRO)/STO(001) substrates using transmission electron microscopy (TEM) and first principles calculations. We chose BM-SFO as a model system to examine the role of the interface on the growth dynamics of a complex oxide thin film. The BM-SFO thin films grew along $b$-axis on both the SRO/STO(001) and STO(001) substrates, while the films are predicted to grow along $a$-axis by the simple epitaxial strain effect. Even though the two substrates have the same in-plane lattice parameters, the thin films grown on both substrates displayed very different heterointerfacial structures and growth processes, suggesting additional key factors that control the growth behaviour of the thin films. We found that these differences in the growth processes of the thin films are attributed to interfacial effects determined by the heterointerfacial structures such as the peculiar transition layer, the different chemical species, and octahedral connectivity.

Results and Discussion

When studying the growth behaviour of metal oxide thin film, it is quite general to consider the epitaxial strain effect first and track down possible factors in detail. This procedure is also applicable to BM thin film despite its complicated crystal structure. Two lattice parameters of orthorhombic BM-SFO, i.e., $b_{pc} = b/4 = 3.900$ Å and $c_{pc} = c/2 = 3.910$ Å, are very similar to that of cubic STO, i.e., $a = 3.905$ Å, while the lattice parameter of $a_{pc} = a/\sqrt{2} = 4.010$ Å of BM-SFO in the $a$-axis direction shows a relatively large lattice mismatch of about 2.7%. The nominal lattice mismatches of BM-SFO in pseudocubic notation with respect to the cubic STO substrate are +2.7%, −0.19%, and +0.08% along the $a$, $b$, and $c$-axes, respectively (Fig. 1). The in-plane area mismatch is also an important parameter for determining the growth characteristics of thin films because the interface is defined by two lattice vectors. Considering the lower surface area mismatch of ~0.11% between the BM-SFO $bc$-plane and the STO(001) plane, $a$-axis growth of BM-SFO on the STO(001) substrate should be energetically much more favourable than $b$-axis growth, which has a larger surface area mismatch of ~2.8%.

The SRO buffer layer has an orthorhombic crystal structure, but the distortion from the cubic structure is very small. Furthermore, the lattice parameter of the pseudocubic unit cell for bulk unstrained SRO ($a = 3.923$ Å) is similar to that of STO, thereby enabling perfectly coherent growth of an epitaxial SRO thin film on the STO(001) substrate to a thickness of about 100 nm. This suggests that the in-plane lattice parameters of the SRO layer are fixed to those of the STO(001) substrate. Thus, when considering only the epitaxial strain effect by lattice mismatch, BM-SFO thin films on both STO(001) and SRO/STO(001) substrates are expected to exhibit the same $a$-axis oriented growth behaviour.

We performed X-ray diffraction (XRD) measurements to confirm the growth orientations of BM-SFO thin films on STO(001) and SRO/STO(001) substrates. Unexpectedly, the BM-SFO thin films revealed the $b$-axis oriented growth behaviour on both substrates rather than the $a$-axis oriented growth behaviour predicted by the epitaxial strain effect. Figure 2a presents an XRD $\theta - 2\theta$ scan of BM-SFO/SRO/STO(001). The pattern shows (010)$_{pc}$ and (020)$_{pc}$ diffraction peaks of BM-SFO near the respective SRO(001)$_{pc}$ and (002)$_{pc}$ peaks. The XRD rocking curve (Fig. 2b) shows a narrow full-width at half-maximum (FWHM) of 0.04°, which suggests good crystallinity of the thin film. Additionally, two-fold superstructure peaks corresponding to (0 1/2 0)$_{pc}$ and (0 3/2 0)$_{pc}$, caused by the ordering of oxygen vacancies, were observed in the XRD pattern. The XRD $\theta - 2\theta$ pattern of SFO/STO(001) also shows the (010)$_{pc}$ and (020)$_{pc}$ diffraction peaks of SFO near the STO(001)$_{pc}$ and (002)$_{pc}$ peaks, and the two-fold superstructure peaks (Fig. 2c). The XRD rocking curve shows a narrow FWHM of 0.02°, which suggests good crystallinity of the thin film (Fig. 2d). The appearance of the (0 1/2 0)$_{pc}$ peaks in these XRD $\theta - 2\theta$ patterns clearly indicates the presence of the BM-SFO phase and its $b$-axis oriented growth on both SRO/ STO(001) and STO(001) substrates$^{20}$.

In order to examine the unexpected growth behaviour of BM-SFO thin films in detail, we characterized the interfacial structures of the BM-SFO thin films using scanning transmission electron microscopy (STEM). Atomic-resolution high-angle annular dark-field (HAADF)-STEM images of SFO films were acquired along the
These images revealed that SFO thin films grown on SRO/STO(001) and STO(001) had quite different atomic structures near the interfaces, even though they all grew into the BM-SFO phase on the substrates having the same in-plane lattice constants. The images of BM-SFO thin films grown on the SRO/STO(001) substrate showed alternate stacking of bright and dark layers along the $b$-axis at the interface, which corresponded...
to FeO$_6$ octahedra and FeO$_4$ tetrahedra in the BM-SFO structure (Fig. 3a). Such a $b$-axis oriented BM-SFO structure appeared over the entire surface of SRO layers. The stacking of octahedral and tetrahedral layers in different orientations, however, was also observed in the BM-SFO film, suggesting that multi-domain growth of the BM-SFO thin film occurred at a few tens-of-nanometres above the interface, possibly due to the strain caused by lattice mismatch between BM-SFO and SRO (see Supplementary Figs. S1 and S2 for the multi-domain structures in BM-SFO and associated functional property of the thin film).26

The atomic arrangement of the BM-SFO thin film at the interface with the SRO layer was investigated using HAADF-STEM imaging and energy dispersive X-ray spectroscopy (EDS) measurements. The Z-contrast nature of HAADF-STEM images clearly revealed the Sr, Fe–O, Ti–O, and Ru–O atomic columns, which are shown as green, red, blue, and yellow spheres in Fig. 3a. The position of the interfaces and atomic columns were accurately verified by EDS element mapping (Fig. 3b). These measurements revealed a remarkably clean interface without misfit dislocation for the BM-SFO films grown on the SRO layer.

In contrast to the film on SRO/STO, the BM-SFO film on the STO substrate showed a quite different interfacial structure. The HAADF-STEM image (Fig. 3c) shows octahedral and tetrahedral layers alternating along the $b$-axis over the entire surface of the STO substrate, indicating $b$-axis growth of the BM-SFO thin film; this is in accord with the XRD measurements (Fig. 2c). The $b$-axis oriented growth of the BM-SFO thin film is identical to that of BM-SFO grown on SRO (Fig. 3a). However, multi-domain growth of BM-SFO, which was observed for BM-SFO/SRO/STO(001), was not observed within the thin film. Additionally, a dark stripe, identified by the green-dashed rectangle in Fig. 3c, was observed at the position where the alternate stacking of bright and dark layers began to form along the $b$-axis. This stripe, which corresponds to a planar defect, formed horizontally and was present even within the films (other types of interfacial structures observed between BM-SFO and STO are shown in Supplementary Fig. S3).

Interestingly, the HAADF-STEM imaging and EDS measurements (Fig. 3c,d) clearly revealed a transition layer between the BM-SFO thin film and the STO substrate. The position of the dark stripe marked by the green-dashed rectangle in Fig. 3c was not located at the interface between the BM-SFO thin film and the STO substrate. The crystal structure of the BM-SFO film on the STO substrate maintained a cubic perovskite-like structure, the same as that of the underlying STO substrate, and changed into the brownmillerite structure mostly with the formation of planar defects within the range of 1–5 nm above the BM-SFO/STO(001) interface. Such planar defects corresponded to Ruddlesden–Popper (RP) planar faults that consisted of a double atomic layers of Fe-O rock-salt structure formed on the surface layer of the perovskite-like SFO with a shear shift of 1/2[110].27,28
Electron energy loss spectrum (EELS) measurements were conducted to identify the transition layer having the unusual perovskite-like SFO structure. Figure 4a shows a HAADF-STEM image of BM-SFO/STO(001) at the interface. The electron-loss near-edge structures (ELNES) of the Ti and Fe $L_{2,3}$-edges and the O $K$-edge were measured across the interface from the STO substrate through the transition layer to the BM-SFO thin film (Fig. 4b–d). The EELS spectra of the STO substrate and the BM-SFO thin film exhibited characteristic Ti (red in Fig. 4b) and Fe $L_{2,3}$-edge (blue in Fig. 4d) structures such as spin-orbit and crystal field splitting, which are identical to those of the stoichiometric compounds.$^{29,30}$ The intensity of the Ti $L_{2,3}$-edge gradually decreased from the STO substrate to the BM-SFO thin film across the transition layer, while that of the Fe $L_{2,3}$-edge increasingly increased. The presence of both Ti and Fe $L_{2,3}$-edges at the boundary between the STO substrate and the transition layer indicated slight cation intermixing between Ti and Fe, i.e. $\text{SrTi}_{1-x}\text{Fe}_x\text{O}_3$, within a few atomic layers. The intensity of the Ti $L_{2,3}$-edge decayed to zero above the intermixing layer, while that of Fe $L_{2,3}$-edge reached nearly its maximum value within the transition layer, indicating that the transition layer above the intermixing layer consisted of the SFO phase. The thickness of the perovskite-like SFO layer varied slightly with position (see Supplementary Fig. S4 for more details).

The valence state of Fe in the perovskite-like SFO phase in the transition layer was further investigated using EELS analysis. The peaks corresponding to the Ti and Fe $L_{2,3}$-edges remained at exactly the same energy loss across the transition layer (Spectra 1–14, Fig. 4b,d), which suggests that the chemical shifts for the Ti and Fe core levels were negligible within the transition layer. Importantly, this indicates that the valence state of Fe was still $3^+$, the same as in the BM-SFO phase, throughout the transition layer despite its perovskite-like crystal structure verified by HAADF-STEM images.

The O $K$-edge ELNES of the perovskite-like SFO layer were also similar to those of the BM-SFO thin film. The STO substrate and the BM-SFO thin film showed their characteristic O $K$-edge structures, as displayed in red and blue in Fig. 4c, respectively.$^{29,30}$ The O $K$-edge spectrum from the STO substrate consists of the first pre-peak and second and third main peaks, in ascending order of energy loss; these are derived from the O $2p$ state hybridized with Ti $3d$, Sr $4d$, and Ti $4s$ states, respectively.$^{31}$ The pre-peak shifted toward lower energy loss, and other two peaks broadened and merged together from the STO substrate to the perovskite-like SFO layer. The O $K$-edge spectra then remained nearly unchanged throughout the transition layer and became similar to that of the BM-SFO phase showing the Fe oxidation state of $3^+$ (oxygen concentration across the interface and the differences in the EELS spectra between SrFeO$_3$ and SrFeO$_{2.5}$ can be found in Supplementary Fig. S5 and Supplementary Information S6). The only difference was that the pre-peak was blunt at the RP planar faults; this was attributed to the different electronic structures of the RP and BM-SFO phases.

These STEM measurements clearly revealed that BM-SFO thin films were grown along $b$-axis on the clean interface of SRO/STO(001) and on STO(001) together with the formation of the transition layer at the heterointerface. Such $b$-axis oriented growth of the BM-SFO thin films on both STO(001) and SRO/STO(001) substrates was opposite to the preferential $a$-axis oriented growth expected on the basis of epitaxial strain energy considering simple lattice mismatches. To account for the discrepancy between the experimental results and the prediction from the conventional epitaxial strain effect, first principles calculations were performed. To identify the energetically favourable interfacial structure, we built atomistic interface models taking into account two possible...
orientations of BM-SFO, i.e., along the \(a\) - and \(b\) -axes (Fig. 5), and compared their heats of formation according to the equation:

\[
\Delta H_f = \frac{1}{A} \left[ E_{\text{slab}} - mE_{\text{substrate}} - nE_{\text{SrFeO}_{2.5}} \right]
\]  

(1)

where \(A\) is the in-plane area of the interface model, \(E_{\text{slab}}\) is its energy including the contribution of the surfaces as well as the interfaces, \(E_{\text{substrate}}\) is the energy of the STO or SRO unit cell, \(E_{\text{SrFeO}_{2.5}}\) is the energy of the \(\text{SrFeO}_{2.5}\) unit cell, and \(m\) and \(n\) denote the number of formula units of the substrate and \(\text{SrFeO}_{2.5}\) included in the interface model, respectively. Because FeO and FeO\(_2\) layers are alternately stacked along the \(b\) -axis in BM-SFO, for the \(b\) -axis orientation of BM-SFO we examined these two structures with the different formula units, namely FeO or FeO\(_2\), at the interface. In our interface models, four layers of SFO lay on top of two layers of STO (or SRO) and their atomic positions were fully relaxed except for the one layer of STO (or SRO) at the bottom, which was considered to be bulk. The in-plane lattice parameters of supercells were fixed to those of STO in every calculation, which reflected the coherent growth of SRO as well as SFO in experiments.

The heats of formation in Eq. (1) indicate the contribution of the surfaces as well as the interface of the atomistic interface models. Both surface and interface energies are crucial to determine the growth behaviour of thin film at the initial growth stage. Thus, the calculations based on our interface models are suitable to compare the stability of each system relatively for the growth of BM-SFO thin films. Table 1 lists the \(\Delta H_f\) for SFO/SRO/STO(001) and SFO/STO(001).

Interestingly, BM-SFO preferred \(b\)-axis growth with FeO\(_2\) units at the interface of the SRO/STO(001) substrate, despite its significant in-plane area mismatch. The reasons for this unexpected result are twofold. First, different chemical elements in the B-site of a BO\(_6\) octahedron result in different interfacial effects for the growth of BM-SFO thin films. The RuO\(_6\) octahedra in SRO and FeO\(_6\) ones in SFO are all slightly rotated and tilted in their bulk forms and can be connected coherently to each other through the formation of FeO\(_6\) octahedral layers at the interface\(^{16,32}\). Such bond coherency between distorted RuO\(_6\) and FeO\(_6\) octahedra may favour the \(b\)-axis growth of BM-SFO, which improves the octahedral connectivity at the interface compared with the formation of alternating FeO\(_6\) octahedral and FeO\(_4\) tetrahedral layers along the surface of the SRO/STO(001) substrate. Therefore, \(b\)-axis oriented growth of a BM-SFO thin film is further stabilized on the SRO/STO(001) substrate. Second, according to previous studies, oxygen is more strongly bound in SRO than in SFO, which makes the oxygen-rich unit, i.e., FeO\(_2\), more favourable at the interface than FeO\(_3\).
Indeed, the interfacial octahedral connectivity plays a significant role in altering the oxygen coordination environment of a thin film at the heterointerface. The corner-connectivity of BO$_x$ octahedra allows an oxide film to mimic the octahedral rotation and tilt, which are transferred across the heteroepitaxial interface from the substrate.$^{12-16}$ These changes in the octahedral framework provide new crystal symmetry and physical properties to the films, such as magnetic and transport properties, which are inaccessible in their bulk phases.$^{12-14,37}$ Meanwhile, we found that the octahedral connectivity functioned as a new type of constraint influencing the growth direction of a thin film having an anisotropic structure. The first deposition layer was a pure FeO$_x$ octahedral layer, which provided better octahedral framework coupling between the film and the substrate and consequently led to the stacking of octahedral and tetrahedral layers in a vertical direction with respect to the substrate.

In contrast to the SRO/STO substrates, calculation results on the STO substrates contradict to the experimental results; a- axis growth of BM-SFO was calculated to be most stable on the STO(001) substrate, opposed to the experimentally observed b- axis growth in Figs. 2c and 3c. Such a discrepancy is possibly attributed to the transition layer observed at the heterointerface of BM-SFO/STO(001), which was not considered for the calculations. The ideal interface model exhibiting an abrupt interface cannot be directly applicable to predict the growth behaviour of BM-SFO thin film on the STO(001) substrate, whereas this model is appropriate for the calculations of BM-SFO/SRO(001) structure with a clean interface (Fig. 3a).

SFO in the transition layer had a perovskite-like crystal structure and an unusual Fe valence state of 3+, as revealed by HAADF-STEM and EELS measurements. The cubic SrFeO$_{2.5}$ phase with a lattice constant of 3.970 Å and Fe-octahedral connectivity STO(001) substrate. Even these two phases can be roughly understood in terms of RP phase were self-generated and served as a buffer layer, resulting in resistive-switching memory. Tailoring the heterointerface could lead to potentially new device applications such as solid-oxide fuel cells and e.g. enhance our understanding of the parameters affecting heterointerfacial structures, which determine the crystal growth behaviour of oxide thin films. The initial growth condition used in pulsed laser deposition (PLD) and molecular beam epitaxy$^{27}$ is in contrast to the SRO/STO substrates, calculation results on the STO substrates contradict to the experimental results for pulsed laser deposition. Therefore, the initial growth condition in the present study might have induced formation of the RP phase and the perovskite-like SFO layer with disordered oxygen vacancies. The influence of such a buffer layer on the growth behaviour of the thin films signifies interfacial effects induced by a specific growth condition.

Conclusion

In conclusion, we demonstrated that the growth process of a BM-SFO thin film can be determined through novel interfacial effects that overwhelm the conventional lattice mismatch argument. These interfacial effects seem to originate from the peculiar transition layer as well as octahedral connectivity and chemical composition at the interface. The BM-SFO thin films grown on two types of substrates, i.e., STO(001) and SRO/STO(001), exhibited b-axis oriented growth with distinct microstructures and interfacial structures, whereas a-axis oriented growth was predicted for both substrates by the simple epitaxial strain effect. Bond coherency between distorted octahedra of RuO$_6$ and FeO$_6$ led to the b-axis orientation of SFO on the SRO/STO(001) substrate, which mimicked the atomic configuration of the SRO buffer layer at the interface. Our concept of octahedral connectivity could be one of the missing links between conventional strain engineering based on simple lattice mismatch and the concept of octahedral-tilt connectivity. Meanwhile, the cubic perovskite-like SrFeO$_{2.5}$ phase with disordered oxygen vacancies and RP phase were self-generated and served as a buffer layer, resulting in b-axis oriented growth of the BM-SFO on the STO(001) substrate. Even these two phases can be roughly understood in terms of octahedral connectivity.

Understanding the growth mechanisms is an essential prerequisite for the manipulation and fabrication of desired oxide heterostructures. Here, we clearly revealed two different growth mechanisms governing the same b-axis oriented growth behavior for BM-SFO/STO(001) and BM-SFO/SRO/STO(001). These findings greatly enhance our understanding of the parameters affecting heterointerfacial structures, which determine the crystal growth behaviour of complex oxide thin films such as BM oxides. More importantly, our results are expected to stimulate research on interfacial effects that could be used to tune the growth process of oxide thin films and the interfacial structure. Controlling the growth process, e.g., the direction of oxygen vacancy channels, by tailoring the heterointerface could lead to potentially new device applications such as solid-oxide fuel cells and resistive-switching memory.

Methods

**BM-SFO thin film growth.** Epitaxial SFO and SRO thin films were grown on the STO(001) substrate using PLD with a KrF excimer laser and commercial (Toshima, Japan) and homemade targets. For the thin film growth, STO substrate was etched using a buffered NH$_4$F-HF (BHF) solution with pH = 4.5. The STO substrate was immersed in the BHF solution for 30 s, rinsed with pure water and ethanol, dried in a nitrogen stream, and then annealed at 1050 °C to obtain a well-defined step-and-terrace STO surface. The 60–70-nm-thick SFO layer was deposited at a substrate temperature of 650 °C, oxygen partial pressure of 10 mTorr, laser fluence of ca. 2.1 J cm$^{-2}$, and a repetition rate of 4 Hz. To prepare the BM-SFO/SRO substrate heterostructure, a 60–70-nm-thick SRO buffer layer was first deposited (substrate temperature: 750 °C; repetition rate: 4 Hz; fluence: ca. 2.5 J cm$^{-2}$) and then the 60–70-nm-thick SFO layer was deposited on top of the SRO layer. Details of the thin film fabrication are reported elsewhere.$^{23,25}$
First principles calculations. First principles calculations were carried out using the Vienna *ab initio* Simulation Package\(^4^0\). We used the projector-augmented wave method to describe electron-ion interactions\(^4^1\). The PBE-GGA functional was used to calculate the exchange-correlation energy\(^4^2\) and an effective on-site energy, \( U_{\text{eff}} \) of 4.3 eV for the Fe 3d states. The cutoff energy for the plane-wave basis set was set to 550 eV. The \( 1 \times 3 \times 1 \) and \( 4 \times 4 \times 1 \)-k-point meshes were used to model the \( a \)-axis and \( b \)-axis growth of the BM-SFO thin films, respectively. Spin-polarized calculations were conducted for all models. The spurious dipole interaction between the slabs was corrected by introducing external dipoles\(^4^3\).

Structural and microstructural characterization. The epitaxial structures of the BM-SFO/SRO/STO(001) and BM-SFO/STO(001) samples were investigated using a high-resolution X-ray diffractometer (Bruker D8). Microstructural analysis of the interfacial structures was performed by TEM. Cross-sectional TEM specimens were prepared by focused ion beam (FIB) milling (Helios 650 FIB, FEI) and thinned by focused Ar-ion milling (NanoMill 1040, Fischione). The atomic-resolution HAADF-STEM images and EDS maps were acquired using a spherical aberration-corrected TEM (JEOL ARM200F, FEI) equipped with a cold field emission gun. Spatially resolved EELS measurements were performed by spectral mapping with a spherical aberration-corrected TEM (Titan Themis 60–300 cubed, FEI) equipped with double Cs correctors, a high-brightness X-FFG module with monochromator, and a GATAN GIF Quantum ERS/966 energy filter. The spectra were recorded using a 300-kV monochromated electron beam. The spectra obtained along the in-plane direction were summed to improve the signal-to-noise ratio. The background was subtracted using the power law. The convergence semi-angle of the monochromated probe was 28.2 mrad, and the acceptance semi-angle for EELS was 40.7 mrad.

Data availability

The datasets generated during and/or analysed during the current study are available from the corresponding author on reasonable request.

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References

1. Goodenough, J. B. Electronic and ionic transport properties and other physical aspects of perovskites. Rev. Prog. Phys. 67, 1915–1993 (2004).
2. Zubko, P., Gariglio, S., Gabay, M., Ghosez, P. & Triscone, J. M. Interface Physics in Complex Oxide Heterostructures. *Annu. Rev. Condens. Ma.* 2, 141–165 (2011).
3. Ohtomo, A. & Hwang, H. Y. A high-mobility electron gas at the LaAlO\(_3\)/SrTiO\(_3\) heterointerface. Nature 427, 423–426 (2004).
4. Nakagawa, N., Hwang, H. Y. & Muller, D. A. Why some interfaces cannot be sharp. *Nat. Mater.* 5, 204–209 (2006).
5. Reyren, N. et al. Superconducting interfaces between insulating oxides. Science 317, 1196–1199 (2007).
6. Mannhart, J. & Schlom, D. G. Oxide Interfaces: An Opportunity for Electronics. *Science* 327, 1607–1611 (2010).
7. Tsukazaki, A. et al. Observation of the fractional quantum Hall effect in an oxide. *Nat. Mater.* 9, 889–893 (2010).
8. Ohtomo, A., Muller, D. A., Grazul, J. L. & Hwang, H. Y. Artificial charge-modalulation in atomic-scale perovskite titanate superlattices. *Nature* 419, 378–380 (2002).
9. Ogawa, N., Satoh, T., Ogimoto, Y. & Miyano, K. Nonlinear optical detection of a ferromagnetic state at the single interface of an antiferromagnetic LaMnO\(_3\)/SrMnO\(_3\) double layer. *Phys. Rev. B* 78, 212409 (2008).
10. Gozar, A. et al. High-temperature interface superconductivity between metallic and insulating copper oxides. *Nature* 455, 782–785 (2008).
11. Izumi, M. et al. Perovskite superlattices as tailored materials of correlated electrons. *Nat. Sci. Eng. B-Solids* 84, 53–57 (2001).
12. Kan, D. et al. Tuning magnetic anisotropy by interfacially engineering the oxygen coordination environment in a transition metal oxide. *Nat. Mater.* 15, 432–438 (2016).
13. Liao, Z. et al. Controlled lateral anisotropy in correlated manganite heterostructures by interface-engineered oxygen octahedral coupling. *Nat. Mater.* 15, 425–432 (2016).
14. O’Sullivan, M. et al. Interface control by chemical and dimensional matching in an oxide heterostructure. *Nat. Chem.* 8, 347–355 (2016).
15. Jeen, H. et al. Topotactic Phase Transformation of the Brownmillerite SrCoO\(_{2.1}\) to the Perovskite SrCoO\(_{2.4}\). *Adv. Mater.* 25, 3651–3656 (2013).
16. Young, J. & Rondinelli, J. M. Crystal structure and electronic properties of bulk and thin film brownmillerite oxides. *Phys. Rev. B* 92, 174111 (2015).
17. Boivin, J. C. & Mairesse, G. Recent material developments in fast oxide ion conductors. *Chem. Mater.* 10, 2870–2888 (1998).
18. Orera, A. & Slater, P. R. New Chemical Systems for Solid Oxide Fuel Cells. *Chem. Mater.* 22, 675–690 (2010).
19. Tambunan, O. T. et al. Resistance switching in epitaxial SrCoO\(_3\), thin films. *Appl. Phys. Lett.* 105, 063507 (2014).
20. Inoue, S. et al. Anisotropic oxygen diffusion at low temperature in perovskite-structure iron oxides. *Nat. Chem.* 2, 213–217 (2010).
21. Ackett, J. E. et al. Combined Experimental and Computational Study of Oxygen Ion Conduction Dynamics in Sr\(_2\)FeO\(_{5}\) Brownmillerite. *Chem. Mater.* 25, 3080–3087 (2013).
22. Mitra, C., Meyer, T., Lee, H. N. & Reboredo, F. A. Oxygen diffusion pathways in brownmillerite SrCoO\(_{2.4}\): Influence of structure and chemical potential. *J. Chem. Phys.* 141, 084710 (2014).
23. Acharya, S. K. et al. Brownmillerite thin films as fast ion conductors for ultimate-performance resistance switching memory. *Nano Lett.* 9, 10502–10510 (2017).
24. Rossell, M. D. et al. Structure of epitaxial Ca\(_2\)Fe\(_2\)O\(_5\) films deposited on different perovskite-type substrates. *J. Appl. Phys.* 95, 5145–5152 (2004).
25. Acharya, S. K. et al. Epitaxial Brownmillerite Oxide Thin Films for Reliable Switching Memory. *Acs Appl. Mater. Inter.* 8, 7902–7911 (2016).
26. Shimakawa, Y. et al. Topotactic Changes in Thin Films of Brownmillerite SrFeO\(_{2.5}\) Grown on SrTiO\(_3\) Substrates to Infinite-Layer Structures Sr\(_2\)FeO\(_{5}\). *Cryst. Growth Des.* 10, 4713–4715 (2010).
27. Suzuki, T., Nishi, Y. & Fujimoto, M. Ruddlesden-Popper planar faults and nanotwins in heteroepitaxial nonstoichiometric barium titanate thin films. *J. Am. Ceram. Soc.* 83, 3185–3193 (2000).
28. Klie, R. F. & Browning, N. D. Atomic scale characterization of vacancy ordering in oxygen conducting membranes. *Microsc. Microanal.* 8, 475–486 (2002).
29. Galakhov, V. R. et al. Valence Band Structure and X-ray Spectra of Oxygen-Deficient Ferrites SrFeOx. J. Phys. Chem. C 114, 5154–5159 (2010).
30. Zhu, G. Z., Radlke, G. & Botton, G. A. Bonding and structure of a reconstructed (001) surface of SrTiO3 from TEM. Nature 490, 384–387 (2012).
31. Seo, H. et al. Band alignment and electronic structure of the anatase TiO2/SrTiO3(001) heterostructure integrated on SrTiO3. Phys. Rev. B 86, 075301 (2012).
32. Haruta, M. et al. Local electronic structure analysis for brownmillerite Ca(Sr)FeO2.5 using site-resolved energy-loss near-edge structures. J. Appl. Phys. 110, 8 (2011).
33. Berger, R. F., Broberg, D. P. & Neaton, J. B. Tuning the electronic structure of SrTiO3/SrFeO3 x superlattices via composition and vacancy control. Apl. Mater. 2, 046101 (2014).
34. Rondinelli, J. M., May, S. J. & Freeland, J. W. Control of octahedral connectivity in perovskite oxide heterostructures: An emerging route to multifunctional materials discovery. MRS Bull. 37, 261–270 (2012).
35. Aso, R., Kan, D., Shimakawa, Y. & Kurata, H. Atomic level observation of octahedral distortions at the perovskite oxide heterointerface. Sci. Rep. 3, 2214 (2013).
36. Kan, D., Aso, R., Kurata, H. & Shimakawa, Y. Phase control of a perovskite transition-metal oxide through oxygen displacement at the heterointerface. Dalton T. 44, 10594–10607 (2015).
37. He, J., Borisevich, A., Kilin, S. V., Pennycook, S. J. & Pantelides, S. T. Control of Octahedral Tilts and Magnetic Properties of Perovskite Oxide Heterostructures by Substrate Symmetry. Phys. Rev. Lett. 105, 222703 (2010).
38. Grenier, J.-C., Ea, N., Pouchard, M. & Hagenmuller, P. Structural Transitions at High-Temperature in Sr2Fe2O5.
39. Hirai, K. et al. Melting of Oxygen Vacancy Order at Oxide-Heterostructure Interface. Acs Appl. Mater. Inter. 9, 30143–30148 (2017).
40. Kresse, G. & Furthmuller, J. Efficient iterative schemes for ab initio total-energy calculations using a plane-wave basis set. Phys. Rev. B 54, 11169–11186 (1996).
41. Blochl, P. E. Projector Augmented-Wave Method. Phys. Rev. B 50, 17993–17997 (1994).
42. Perdew, J. P., Burke, K. & Ernzerhof, M. Generalized gradient approximation made simple. Phys. Rev. Lett. 77, 3865–3868 (1996).
43. Bengtsson, L. Dipole correction for surface supercell calculations. Phys. Rev. B 59, 12301–12304 (1999).

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Author contributions
J.J. was primary author performing TEM experiments and the majority of analysis. V.R.N. and S.K.A. performed crystal growth using PLD and XRD measurements of BM-SFO thin films. Y.K., S.Y., and S.L. performed first principles calculations. Y.K. helped to acquire atomic-resolution EDS mapping data. H.B., S.H., M.K., and C.U.J. supervised the whole process including experiments and analysis. The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

Competing interests
The authors declare no competing interests.

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Correspondence and requests for materials should be addressed to M.K. or C.U.J.

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