RNA-Seq Reveals the Effect of Ethylene on the Volatile Organic Components (VOCs) of Cavendish Banana at Different Post-harvesting Stages

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Abstract

Aroma serves as one of the decisive factors influencing the value of banana commodities. Most of characteristic volatile organic components (VOCs) are formed during post-harvesting. However, the changing of VOCs of banana at different post-harvesting stages remain ambiguous. In this study, the VOCs of Cavendish banana for the four typical post-harvesting stages (green stage/half of yellow stage/yellow ripening stage/over ripening stage) are clarified using headspace solid phase micro-extraction (HS-SPME), combined with gas chromatography-mass spectrometry (GC–MS). The results inferred that the relative content of branched-chain esters such as acetate and butyrate, which form the main contributors of aroma in bananas, is higher in the T2 and T3 stages. Further, RNA-Seq technology was employed to clarify the formation mechanism of banana aroma in the post-harvesting stage. The MaTGL4 gene of the linoleic acid metabolism pathway and the MaBCAT3 and MaBCAT5 genes of the valine, leucine and isoleucine degradation pathway in banana suggest the expression is active late in the ripening stage, and the upregulated expression of these genes is analogous to the formation of aroma components such as branched-chain esters and hexenal. The above results not only provide baseline data on the differences in physical and chemical properties of VOCs in various post-harvesting stages of banana production, but also provide theoretical guidance facilitating the subsequent improvement of the commercial value of bananas through genetic improvement.

Keywords Banana · Candidate genes · Fruit ripening · Transcriptome · Volatile aroma components

Abbreviations

VOCs Volatile organic components
HS-SPME Headspace solid phase micro-extraction
GC–MS Gas chromatography-mass

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pivotal economic stature (Rustagi et al. 2015; Wei et al. 2017; Li et al. 2020). Aroma is one of the most appreciated characteristics of any fruit and is seen as a profound determining factor for its acceptability by consumers (Selli et al. 2012; Dou et al. 2020). As a vital economic fruit, banana is rich in VOCs, which include esters, alcohols, phenols, ketones, aldehydes, ethers, and terpenes (Boudhrioua et al. 2003). VOCs are highly accountable for the aromatic flavor of banana. Aroma constituents of banana (Facundo et al. 2012; Dong et al. 2014; Bugaud and Alter 2016; Saha et al. 2018) have been extensively studied. Approximately 250 volatile compounds have been detected in banana at different stages of ripeness and process (Jordan et al. 2001; Pino and Febles 2013). Generally, the major volatile compounds are classified as esters, alcohols, and ketones. Furthermore, several reports suggest that acetates (especially isoamyl acetate and isobutyl acetate) and butyrate are highly influential for the development of aroma of a fully ripe banana fruit (Marriott and Palmer 1980; Hadi et al. 2013). Nogueira et al. (2003) investigated the composition of volatiles found in banana cultivars (Dwarf Cavendish, Giant Cavendish, Robusta and Williams) from Madeira Island. It was deduced that the ester (57.2–89.8 mg/kg) and alcoholic fractions (19.0–47.7 mg/kg) of the volatiles observed appeared to play a decisive role in the aroma characteristics of banana fruit. Experiments using GC–olfactometry (Gas Chromatography Olfactometry) strongly advocated for 3-methylbutylacetate, 3-methyl-butyl butanoate, and 3-methylbutyl-3-methyl butanoate to be the key components responsible for banana’s fruity odor (Facundo et al. 2012).

Ripening tends to be the amalgamation of processes that initiate at the latter stages of growth and proceed up to the early stages of senescence, resulting in the genesis of characteristic properties of the ripe fruit-like flavor, taste, texture, color. Ripeness being the determining parameter for substrate availability in the plant system, in turn makes biogenesis of fruit aroma evidently dependent on the ripeness of the material. At varied stages of ripening, Zhu et al. (2010) investigated the transformations of the volatile components of banana fruit, simultaneously exploring the relationships between volatile production and other physiological events occurring during fruit ripening. Moreover, it was reported that changes in the production of total volatile compounds by banana fruit during the process of storage were closely affiliated to the event of fruit ripening and the ripening-associated parameters, such as fruit ethylene production, respiration rate, firmness, color, soluble solids content, and starch content (Zhu et al. 2010). Also, some volatiles come across as particularly good indicators of the ripening stage of bananas, such as hexanal and (E)-2-hexenal, as well as isobutyl acetate and isopentyl acetate, which are characteristic of ripe and ripening bananas (Mayr et al. 2003).

It is evidently known that both fatty acid and amino acid pathways can produce aldehyde, followed by reduction to alcohol by alcohol dehydrogenase (ADH) (Speirs et al. 1998), and finally generate esters catalyzed by alcohol acyltransferase (AAT) (Beekwilder et al. 2004; Souleyre et al. 2014). Studies have proven that leucine, isoleucine, valine, and alanine are the major biosynthetic precursors of branched-chain esters in banana, strawberry and several other fruits that are closely related to the formation of esters in fruits (Wyllie and Fellman 2000; Pérez et al. 2002). Considering the primary aroma of several fruits (Pérez et al. 2002), these compounds either act as major contributors or behave as determining factors in some cases. Due to the development of ‘Omics technology’ (universal detectors of complete genetic or molecular profile of any organism, classified into genomics, transcriptomics, etc.,) in recent years, several genes closely associated with the synthesizing of banana volatile substances have been isolated and identified. For example, the genes encoding aroma synthesis in banana fruit have been partially identified, including O-methyltransferases 1 (OMT1), branched-chain amino acid transaminase 1 (BCAT1), pyruvate dehydrogenase complex (PDC), basic leucine zipper 4 (bZIP4), basic leucine zipper 5 (bZIP5), MaADH, and BanAAT (Yang et al. 2011; Guo et al. 2018). Some studies conducted by Asif et al. (2014) suggested that the expression of MaADH demonstrated more than twofold upregulation in the case of a ripe fruit as compared to MaOMT, MaBCAT1, and MaAAT genes and some other unripe fruit. As mentioned above, previous studies have described the major contributors of banana volatile aroma. This paper undertakes a preliminary exploration at the genetic level to further divulge in comprehending the mechanism of banana aroma substances formation. However, the changing rules and mechanisms of banana VOCs at different post-harvesting stages remain vague and cryptic.

In this study, firstly, HS-SPME coupled with GC–MS technology was employed to evaluate the changing rules of banana VOCs in different post-harvesting stages. Geared towards clarifying the formation mechanism of banana VOCs in various post-harvesting stages, conduction of RNA-Seq sequencing analysis for four typical post-harvesting stages of banana fruits was undertaken. Several studies advocate that the relative content of branched-chain esters and branched-chain alcohols in the late ripening stage of banana fruits shoots up significantly, and three certain genes related to lipid and amino acid metabolism are actively expressed in the late banana ripening stage, noticeably predicting their impending correlation with the formation of aroma substances in the later stages of banana ripening. These findings serve as a robust stand on the basis of which further in-depth analysis of the metabolic mechanism of aromatic substances during banana ripening at the molecular level could be undertaken.
Materials and Methods

Plant Materials and Treatment

The cultivar of mature, green Cavendish banana fruit (Musa acuminata, AAA group, cv. Cavendish) at the 75–80% plump stage were harvested from the plantation (23° 75′ N, 114° 25′ E) of Institute of Fruit Tree Research, Guangdong Academy of Agricultural Sciences. To rule out the impact of different harvest seasons on fruit aroma, banana bunches were harvested at the same time and under the same physiological maturity stage (caliber 33 ± 1 mm, measured in the middle finger of outer whorl of the second hand from the distal end of the bunch). The second hand from the proximal end of each bunch was selected (around 60 kg of banana), then dipped in 1.0 µL/L ethephon for 1 min. The fruit was naturally dried and then stored in a polyethylene film bag at 18 °C for 36 h. Post that, the bag was opened but continued to be placed at 18 °C until the banana fruit completed the whole post-harvesting process. According to the peel color measurement results and phenotype changes, it is determined that the 1st day of post-harvesting without ethephon treatment is the green stage (CK), the 3rd day after ethephon treatment is half of yellow stage (T1), the 6th day after ethephon treatment is yellow ripening stage (T2), and the 9th day after ethephon treatment is over ripening stage (T3). The fruits were peeled, and their pulp was sliced, frozen in liquid nitrogen and stored at −80 °C for subsequent transcriptome analysis. Three independent measurements were conducted on randomly selected fruits as replicates for each ripening stages.

Color of Banana Peels

The color of banana fruit was analyzed and measured according to the method we reported previously (Dou et al. 2020). The colorimetric coordinates of fresh banana peels were measured using a Minolta Chroma Meter CR 400 (color space CIE L*, a*, b* and h*). To avoid experimental errors, three banana fruits were randomly tested at each time point, and three fixed parts were tested for each fruit.

Volatile Analysis

The VOCs of bananas were determined according to the method previously reported in this study (Dou et al. 2020). The 50/30 µm Divinylbenzene/Carboxen/Polydimethylsiloxane (DVB/CAR/PDMS) fibers extraction head manufactured by Supelco, Bellefonte, Pa., U.S.A was used for the test. The extraction head was aged at 250 °C for 2 h. 20 g of homogenate was sealed in a 50 ml headspace vial. Solid-Phase Micro extraction (SPME) was carried out for 30 min and desorbed at 220 °C for 3 min. Samples were analyzed using a Finnigan Trace GC–MS instrument. The GC–MS conditions were: DB-1 column (length 30 m, inner diameter 0.25 mm, liquid film thickness 0.1 µm), column temperature 40 °C for 1 min, then increased by 2 °C/min to 60 °C (for 2 min), and further increased by 10 °C/min to 180 °C (for 5 min). The inlet temperature was 220 °C. Carrier gas He (99.99%) was provided at a flow rate of 1.0 mL/min. The EI ion source has a temperature of 170 °C, an electron energy of 70 eV, and a photomultiplier tube voltage of 350 V was employed. The mass scan range was 35–335 amu. The total ion chromatogram of the collected aroma substances was determined using the NIST 11 and Wiley mass spectral libraries to determine the chemical composition of the aroma components. The relative content of each component was obtained using the peak area normalization method for quantification.

RNA Extraction and RNA-Sequencing

Total RNA derived from banana pulp was isolated using a plant RNA extraction kit (Code No. 9767, TaKaRa, Dalian, China), and first strand cDNA was synthesized from 1 µg of total RNA using a PrimeScript RT Reagent Kit, according to the manufacturer’s instructions (TaKaRa). Triplicates were made at each time point, and each biological replicate randomly selected a mixed sample of three fruit pulps. RNA preparation, library construction, and sequencing on the BGISEQ-500 platform of PE 100 with 30 million reads per sample was performed at BGI (http://www.genomics.org.cn, BGI, Shenzhen, China). Gene expression levels were quantified using the RSEM software package (Li and Dewey 2011). The DEGseq method was utilized to screen differentially expressed genes (DEGs) between groups as previously described in this study (Huang et al. 2018). Gene Ontology (GO) and pathway annotation and enrichment analyses were based on the GO Database (http://www.geneontology.org/) and the KEGG pathway database (http://www.genome.jp/kegg/), respectively. Statistical analysis was performed and DEGs were selected with the criteria of a fold change ≥ 2 and adjusted p value ≤ 0.001, as described previously (Huang et al. 2018).

Quantitative Real-Time (RT-qPCR) PCR Analysis

To further validate the reliability of this transcriptomic data, total RNA derived from banana pulp at four different time points was isolated using a plant RNA extraction kit (Code No. 9767, TaKaRa, Dalian, China), and the first strand cDNA was synthesized from 1 µg of total RNA using a PrimeScript RT Reagent Kit, according to the manufacturer’s instructions (TaKaRa). The resulting cDNAs were then employed as templates for qRT-PCR. 8 genes were
assayed by RT-qPCR with specific primers as shown in File S3. The PCR reaction consisted of 10 µL of 2 × SYBR Green PCR Master Mix (TaKaRa), 200 nM primers, and 2 µL of 1:40-diluted template cDNA in a total volume of 20 µL. The RT-qPCR program included an initial denaturation step at 95 °C for 10 s, followed by 40 cycles of 5 s at 95 °C, and annealing and extension at 60 °C for 20 s. The MaACT1 gene (File S3) was selected as a reference gene base on our previous report (Dou et al. 2016). The relative expression levels of the amplified products were calculated based on the comparative threshold cycle method (Li and Dewey 2011). Transcript abundance of the 8 candidate genes was normalized against the reference gene, and the three technical replicates were performed for each gene.

Results and Discussion

Determination of Typical Post-harvesting Stage of Banana Fruit

The banana peel gradually turned yellow, and subsequently on the 9th day of storage, the banana peel demonstrated small sesame spots (Fig. 1a). Afterwards, the fruit rotted and lost its commercial value. In order to determine the time point of sampling at different post-harvesting stages of bananas, we measured the color of banana peels, as shown in Fig. 1b. With the extension of storage time, the red saturation a* value proclaimed a recent upward trend, but h* portrayed the gradual decrease trend, inferring that the banana fruit gradually completed the post-harvesting process; both L* and b* initially showed an elevating trend but then led to a decline, and both reached the maximum value on the 6th day of storage, indicating that the banana fruit had completed the post-mature date since the period after 6 days is the over-mature stage.

Previous studies demonstrate that volatile components of banana fruits are affected by genotype (Aurore et al. 2011; Pontes et al. 2012; Bugaud and Alter 2016), cultural practice (Brat et al. 2004; Dong et al. 2014), stages of ripening (Zhu et al. 2018), processing conditions (Wang et al. 2007), harvest season (Shivashankara et al. 2017; Dou et al. 2020) and storage conditions (Boudhrioua et al. 2003), along with several other factors. These factors tend to regulate the quantitative and qualitative composition of the volatile profile, which in most cases determines the character and perception of quality of the fruits (Brat et al. 2004). Therefore, exploring the changing rules and mechanisms of aroma components of

![Fig. 1](image-url)
banana fruits at different post-harvesting stages proves to be an extensively vital step. Whereas, determining the typical post-harvesting stage is the crucial link of this study.

**Variation of Banana Volatile Components at Different Post-harvesting Stages**

Combining dynamic headspace solid-phase micro-extraction with one-dimensional gas chromatography–mass spectrometry (GC–MS) serves as a powerful tool to differentiate banana cultivars based on their volatile metabolite profile (Dou et al. 2020). To further explore the changing rules of the aroma components in banana fruits at various post-harvesting stages, the above four post-harvesting periods were selected, HP-SPME and GC–MS were adopted to analyze the volatile compounds, and the compounds were identified based on the MS data. As shown in Table 1, in the post-harvesting stage of bananas, a total of 87 volatile components were detected, which mainly include alcohol, ester, aldehyde, acid, hydrocarbon, ketone and several other compounds.

The main aroma components of banana are: Acetic acid; 2-methylpropyl ester/1-Butanol; 3-methyl-; acetate/Butanoic acid; 3-methylbutyl ester/Butanoic acid; 3-methyl-; and 3-methylbutyl ester, which are mainly formed by fatty acid metabolism of ester acyl-CoA and the corresponding amino acid metabolite isobutanol or isoamyl alcohol (Jordan et al. 2001; Bugaud and Alter 2016). As shown in Table 1, the ester components in this study were mainly detected in the T2 and T3 stages, with relative content of 8.45% and 26.60%, respectively. The ester content detected in the CK and T1 stages were only about 0.14% and 0.47%, respectively. Also, the esters detected in the T2 and T3 stages mainly turn out to be acetate (2.67%, 18.03%) and butyrate (4.3% and 5.72%). They are predominantly the main aroma substances of banana flavor, while the acetate can emit an attractively strong scent (Wyllie and Fellman 2000) even when present in low concentration. The biosynthetic pathway for their formation in ripening climacteric fruits, such as apples, bananas, apricots, is well-established. Several investigations demonstrate that the volatile esters that are produced by fruits are not only determined by AAT specificity but also by the substrate availability, which is highly dependent on the ripening stage of the fruit (Beekwilder et al. 2004).

The relative content of alcohol compounds in the four different post-harvesting stages is 10.75%, 12.60%, 9.13%, and 18.94%, respectively, and the composition of the alcohol component tends to be significantly distinct in each stage, among which the alcohol detected in the CK and T1 stages is mainly sitosterol, while the alcohol traces in T2 and T3 phases are mainly 1-Butanol; 3-methyl-, 1-Propanol; 2-methyl-, (S)-(+-)-2-Pentanol and 1,2-Propanediol; 3-methoxy- and other branched short fatty alcohol (Table 1). However, alcohols remain less crucial as flavor molecules due to their high odor thresholds as compared to their aldehyde homologues (Marriott and Palmer 1980). Branched-chain fatty alcohols seem to be closely related to amino acid catabolism (Xu et al. 2017).

Both alcohols and minor carbonyl compounds are referred to as green and woody notes. (E) and (Z)-hept-4-en-2-one have fatty-green and sweet notes (Miranda et al. 2001; Pino et al. 2017). As shown in Table 1, the detected acids are mainly medium and long-chain fatty acids. The fatty acids detected in the CK and T1 phases are relatively high, at 58.20% and 63.46%, respectively; the contents of the T2 and T3 phases are relatively low, at 37.22% and 17.82%, respectively. The main acid component is identified as n-Hexadecanoic acid, whose ratios in CK/T1/T2/T3 stage are 29.22%, 24.03%, 22.00%, and 9.77%, respectively. The detected aldehydes are mainly Hexanal and 2-Hexenal, (E)-, whose relative content is higher in the T2 and T3 stages, but present in minute traces in the CK and T1 stages. The formation of Hexanal and 2-Hexenal, (E)- comes from the degradation of fatty acids (Yvon and Rijnen 2001; Ardö 2005). The detected hydrocarbons are mainly straight-chain alkanes or olefins. There are fewer types of hydrocarbon components detected in the T2 and T3 stages, and the content of the hydrocarbon components in the T3 stage is at least 1.82%. In addition, some ethers, amides, and benzene rings in different post-harvesting stages of bananas were detected as well.

**Differential Expression of Genes During Post-harvesting of Bananas**

The post-harvesting process of banana fruits marks the involvement of many genetic differences (Asif et al. 2014). In order to explore the formation mechanism of banana aroma substances at different post-harvesting stages and to screen and analyze candidate genes involved in the formation of banana aroma substances, we used CK as the comparison group for reference and T1, T2, and T3 stages as different post-harvesting stages of bananas. The RNAseq high-throughput sequencing was performed separately. The sequencing instrument was: illumina Hiseq 2500, and the sequencing strategy was double-end reads 150 bp, that is, PE 150 sequencing. The DESeq2 software was utilized to analyze the difference between the two groups. The screening conditions for the differential genes were: FDR value less than 0.05 and difference multiple ≥ 2. As shown in Fig. 2, compared to CK, there are 4277 genes upregulated and 8,412 genes downregulated in T1 phase; 4587 genes upregulated and 8765 genes downregulated in T2 phase; 4292 genes upregulated and 8923 genes downregulated in T3 phase; 9630 genes have differential expression changes.
| Retention time | Volatile compounds | Relative content (%) |
|---------------|-------------------|----------------------|
|               |                   | CK       | T1       | T2       | T3       |
| **Alcohols**  |                   |          |          |          |          |
| 2.05          | 1-Propanol; 2-methyl- | –   | –   | –   | 4.91 |
| 2.60          | (S)-( +)-2-Pentanol  | –   | –   | 1.39 | 1.92 |
| 2.99          | 1-Butanol; 3-methyl- | –   | –   | 3.53 | 6.85 |
| 4.01          | 1,2-Propanediol; 3-methoxy- | 1.18 | 0.75 | 2.43 | 1.37 |
| 5.00          | 1-Hexanol          | 0.08  | 0.41  | 0.29  | 0.21 |
| 8.83          | 4-Nonanol          | 1.06  | 1.18  | 1.50  | 1.52 |
| 10.80         | 3-Pentanol; 2,4-dimethyl- | –   | –   | –   | 0.26 |
| 12.94         | 1-Heptanol; 2-propyl- | 0.54 | –   | –   | –   |
| 22.10         | n-Tetracosanol-1   | 0.53  | 0.80  | –   | –   |
| 25.45         | 1-Heptacosanol     | 0.18  | 0.29  | –   | –   |
| 25.84         | .gamma.-Sitosterol | 0.00  | 0.00  | –   | 1.89 |
| 34.88         | Ergost-5-en-3-ol; (3.beta.)- | 0.77 | 1.04 | –   | –   |
| 35.57         | Stigmasterol       | 0.74  | 0.90  | –   | –   |
| 37.33         | .gamma.-Sitosterol | 5.67  | 7.25  | –   | –   |
| **Total**     |                   | 10.75  | 12.60  | 9.13  | 18.94  |
| **Esters**    |                   |          |          |          |          |
| 3.48          | Acetic acid; 2-methylpropyl ester | –   | –   | 0.42 | 5.20 |
| 4.08          | Acetic acid; butyl ester | –   | –   | –   | 0.87 |
| 4.63          | 2-Pentanol; acetate | –   | –   | 0.77 | 1.26 |
| 5.13          | 1-Butanol; 3-methyl-; acetate | –   | –   | 1.48 | 10.70 |
| 6.51          | Butanoic acid; 2-methylpropyl ester | –   | –   | 0.36 | 0.91 |
| 7.41          | Butanoic acid; 3-methyl-; 2-methylpropyl ester | –   | –   | –   | 0.38 |
| 7.50          | Propanoic acid; 2-methyl-; 3-methylbutyl ester | –   | –   | –   | 0.63 |
| 7.62          | 4-Hexen-1-ol; (4E)-; acetate | –   | –   | –   | 0.28 |
| 7.68          | Butanoic acid; 1-methylbutyl ester | –   | –   | 0.74 | 0.33 |
| 7.80          | Propanoic acid; 2-methyl-; 1-methylethyl ester | –   | –   | 0.00 | 0.30 |
| 8.25          | Butanoic acid; 3-methylbutyl ester | –   | –   | 2.61 | 3.44 |
| 9.08          | Butanoic acid; 3-methyl-; 3-methylbutyl ester | –   | –   | 1.69 | 2.28 |
| 10.72         | Butyric acid; 4-pentadecylester | –   | –   | 0.37 | –   |
| 19.51         | Hexadecanoic acid; methyl ester | 0.14 | 0.47 | –   | –   |
| **Total**     |                   | 0.14   | 0.47   | 8.45  | 26.60  |
| **Acids**     |                   |          |          |          |          |
| 3.61          | Hexanoic acid     | –   | –   | –   | 1.43 |
| 4.44          | Butanoic acid; 3-methyl- | 0.15 | 0.39 | 0.52 | 0.87 |
| 6.85          | Hexanoic acid     | 0.20  | 0.23  | –   | –   |
| 10.06         | Octanoic acid     | 0.53  | 0.00  | –   | –   |
| 11.48         | 3-Nonenoic acid   | 0.36  | –   | –   | –   |
| 14.54         | Azelaic acid      | 8.55  | 0.50  | –   | –   |
| 15.16         | 12-Hydroxydodecanoic acid | 0.92 | –   | –   | –   |
| 17.73         | Tetradecanoic acid | 0.42 | 0.45 | 0.35 | –   |
| 18.60         | (E)-13-Docosenoic acid | 0.24 | 4.07 | 0.59 | 0.53 |
| 18.82         | Pentadecanoic acid | 0.70 | 0.81 | –   | –   |
| 19.69         | cis-9-Hexadecenoic acid | 4.93 | 6.08 | 1.05 | 0.67 |
| 19.98         | n-Hexadecanoic acid | 29.22 | 24.03 | 22.00 | 9.77 |
| 20.64         | Oleic acid        | 0.09  | 1.50  | 1.63 | –   |
| 20.84         | Heptadecanoic acid | 0.28 | 0.41 | –   | –   |
| 21.52         | 9,12-Octadecenoic acid (Z,Z)- | 0.25 | 10.56 | 1.73 | 0.85 |
| 21.57         | 9,12,15-Octadecatrienoic acid (Z,Z,Z)- | –   | –   | 5.94 | 2.87 |
Table 1 (continued)

| Retention time | Volatile compounds                          | Relative content (%) |     |
|----------------|---------------------------------------------|---------------------|-----|
|                |                                             | CK      | T1  | T2  | T3  |     |
| 21.64          | cis-Vaccenic acid                           | 6.59    | –   | –   | –   |     |
| 21.68          | cis-10-Heptadecenoic acid                   | –       | 10.00 | –   | –   |     |
| 21.80          | Octadecanoic acid                           | 3.45    | 4.44 | 3.39 | 0.83|     |
| 23.70          | 17-Octadecynoic acid                        | 1.32    | –   | –   | –   |     |
| Total          |                                             | 58.20   | 63.46 | 37.22 | 17.82|     |

**Aldehyde**

| Retention time | Volatile compounds                          | Relative content (%) |
|----------------|---------------------------------------------|---------------------|
| 3.88           | Hexanal                                     | 0.65                |
| 4.74           | 2-Hexenal; (E)-                             | 0.08                |
| 5.34           | Hexanal; 3-methyl-                          | 0.08                |
| 9.85           | 2,6-Nonadienal; (E,Z)-                     | 0.25                |
| 9.96           | 2-Nonenal; (E)-                            | 0.22                |
| 11.99          | 2,4-Decadienal                             | 0.38                |
| 12.33          | 2,4-Decadienal; (E,E)-                     | 0.42                |
| **Total**      |                                             | 2.01                |

**Ketones**

| Retention time | Volatile compounds                          | Relative content (%) |
|----------------|---------------------------------------------|---------------------|
| 2.48           | 2-Pentanone                                 | –                   |
| Total          |                                             | 0                   |

**Hydrocarbons**

| Retention time | Volatile compounds                          | Relative content (%) |
|----------------|---------------------------------------------|---------------------|
| 2.35           | Pentane; 2,3-dimethyl-                      | 0.25                |
| 2.64           | 2-Hexene; 4-methyl-; (E)-                   | 0.21                |
| 10.58          | Dodecane                                    | 2.14                |
| 12.75          | Hexane; 3-ethyl-3-methyl-                   | 0.84                |
| 13.32          | 1-Tridecene                                 | 0.65                |
| 13.43          | Tetradecane                                 | 2.57                |
| 15.60          | Hexadecane; 2,6,10,14-tetramethyl-          | 0.16                |
| 15.87          | 1-Pentadecene                               | 0.93                |
| 15.96          | Hexadecane                                  | 0.33                |
| 17.12          | Eicosane                                    | 0.17                |
| 18.15          | 1-Octadecene                                | 0.84                |
| 20.16          | 1-Nonadecene                                | 0.05                |
| 23.03          | Pentacosane                                 | 0.29                |
| 23.88          | Tetracosane                                 | 0.38                |
| 27.45          | Hexatriacontane                             | 0.39                |
| 27.63          | Squalene                                    | 0.15                |
| **Total**      |                                             | 10.50               |

**Others**

| Retention time | Volatile compounds                          | Relative content (%) |
|----------------|---------------------------------------------|---------------------|
| 2.20           | Butane; 1-ethoxy-                           | –                   |
| 2.20           | Propane; 1-ethoxy-2-methyl-                 | 2.27                |
| 3.28           | Pentane; 1-ethoxy-                          | 5.82                |
| 3.43           | Toluene                                     | 0.90                |
| 4.90           | Ethylbenzene                                | 0.44                |
| 14.31          | 2,5-di-tert-Butyl-1,4-benzoquinone          | 0.22                |
| 15.27          | Benzene; 1,2,3-trimethoxy-5-(2-propenyl)-   | –                   |
| 16.59          | 2-n-Octylfuran                              | 0.11                |
| 19.36          | 7,9-Di-tert-butyl-1-oxaspiro(4,5)deca-6,9-die| 0.25                |
| 19.85          | Dibutyl phthalate                           | 3.42                |
| 21.97          | Hexadecanamide                              | 0.72                |
| 23.56          | 9-Octadecanamide; (Z)-                      | 1.59                |
| 23.77          | Octadecanamide                              | 0.47                |
from CK during T1, T2, and T3 periods; complete details are given in File S1.

KEGG Functional Enrichment Analysis of DEGs

The most crucial aroma compounds include, among others, amino acid-derived compounds, lipid-derived compounds, phenolic derivatives, and mono- and sesquiterpenes (Schwab et al. 2008). In this study, through the KEGG enrichment analysis of 9630 genes that were differentially expressed in the T1, T2, and T3 periods with CK, it was determined that when $P \leq 0.05$. The identified DEGs in profile 21 were annotated with 21 biological processes, 13 cellular components and 12 molecular functions in GO categories (Figure S1). Among the biological processes, a great number of DEGs were focused on categories of “metabolic process”, “cellular process”, “single-organism process”, “biological regulation”, “regulation of biological process”, “localization”, “response to stimulus” and “cellular component organization or biogenesis”; The terms of “cell”, “cell part”, “organelle”, “macromolecular complex”, “organelle part”, “membrane” and “membrane part” were the dominant groups in the cellular component; “binding”, “catalytic activity” and “transporter activity” were the representative groups in the molecular function. Furthermore, the significantly DEGs were blasted to the KEGG database to analyze their biological pathways. Among 111 of the enriched KEGG pathways, the most significant ones ($Q \leq 0.05$) consist of “Oxidative phosphorylation”, “Glycolysis/Gluconeogenesis”, “Pyruvate metabolism”, “Carbon fixation in photosynthetic organisms” and “Citrate cycle (TCA cycle)” (Figure S2), followed by others, such as “Valine, leucine and isoleucine degradation”, “Propanoate metabolism”, “SNARE interactions in vesicular transport”, “Pantothenate and CoA biosynthesis”, and “Gluconolactone biosynthesis”.

The linoleic acid metabolism pathway was not only enriched in the top 20, but was significantly enriched as well (Fig. 3a). By extracting the genes upregulated with the CK stage, and then doing KEGG enrichment analysis, when $P \leq 0.05$, it was found that among the top 20 pathways of enrichment degree, the enrichment of valine, leucine, and isoleucine degradation pathway obtained a place in the top 20 and proved to be significantly enriched (Fig. 3b). Also, these two pathways are the main pathways of aroma synthesis (Yvon and Rijnen 2001; Dudareva et al. 2013). The above research indicates that the formation of aroma substances during the post-harvesting process of bananas could establish the possible relationship of this formation to the two metabolic pathways of linoleic acid metabolism and valine, leucine, and isoleucine degradation.

DEGs Related to Valine, Leucine, and Isoleucine Degradation and Linoleic Acid Metabolism

The LOX metabolic pathway is mainly responsible for the degradation of unsaturated fatty acids, one of the most prominent examples being (E)-2-hexenal with its intense fresh green, green apple-like sensory properties that acts as a key enhancer of the flavor of many fruits and vegetables (Dudareva et al. 2013). Investigations suggest that at least four enzymes are involved in the biosynthetic pathway leading to their formation: lipoxygenase (LOX), hydroperoxide lyase (HPL), 3Z,2E-enal isomerase, and alcohol dehydrogenase (ADH).

In the linoleic acid degradation pathway, the expression levels of LOX gene and ADH gene were found to be active in T2 and T3 stages, which can presumably be related to the increase in the content of Hexanal and 2-Hexenal; (E)- at the ripening stage of banana fruits. Dulermo et al. suggested that TGL genes are key performers in aroma synthesis (Dulermo...
et al. 2013). In this study, heat maps of genes enriched in linoleic acid metabolism revealed that 15 genes were enriched in this pathway. Further analysis revealed that *MaTGL4* (GSMUA-Achr3G14540-001) was directly involved in the linoleic acid metabolism pathway, and the expression of *MaTGL4* was significant at the later stage of ripening (Fig. 4), pointing out clearly that *MaTGL4* gene may also be involved in regulating the synthesis of banana aroma.

The direct metabolic pathway of amino acids leads to the formation of aliphatic, branched, and aromatic alcohols, aldehydes, and ketones as well as acids and esters (Dudareva et al. 2013). The major aroma compounds derived from branched-chain and aromatic amino acids and methionine, valine, leucine, and isoleucine are the main branched-chain amino acids (Yvon and Rijnen 2001). When leucine is the initial free amino acid, 3-methylbutanol and 3-methylbutanoic acid are formed, whereas isoleucine would be the precursor for 2-methylbutanol and 2-methylbutanoic acid. The branched acids may further esterify with alcohols to form the corresponding esters (Yvon and Rijnen 2001). In this paper, a heat map of genes enriched in the valine, leucine, and isoleucine degradation pathways was performed, where it was found that *MaBCAT3* (GSMUA-Achr3G15170-001) and *MaBCAT5* (GSMUA-Achr3G15180-001) are directly involved in the valine, leucine, and isoleucine degradation pathways, and they all belong to the branched-chain amino acid aminotransferase (*BCAT*) gene family. Guo et al. showed that branched-chain amino acid transaminase 1 (*BCAT1*) gene is related to the synthesis of aromatic substances (Miranda et al. 2001). The development of a pleasant aroma during fruit ripening is a trait of the metabolism of branched amino acids such as leucine with deamination and decarboxylation (Dou et al. 2020). The expression of *MaBCAT3* and *MaBCAT5* genes screened in this study were found to be active in the late ripening stage in the banana fruit (Fig. 5), indicating that these two genes may dominantly regulate the formation of aroma substances in the late ripening stage of bananas.

Volatile metabolites, such as aliphatic saturated or unsaturated alcohols, aldehydes, ketones, acids, esters, or lactones, derived from fatty acids as non-volatile precursors remain a crucial parameter for the flavor of most fruits and vegetables. β-Oxidation is believed to be the main pathway for the degradation of straight-chain fatty acids in intact fruits and vegetables (Goepfert and Poirier 2007). In β-oxidation, the acyl-CoA derivatives of fatty acids are degraded stepwise so that in every reaction cycle the chain is shortened by two C atoms. Finally, acetic acid or butyric acid is generated, and the main aroma is catalyzed by AAT.
to produce branched-chain acetate or butyrate (Baker et al. 2006; Bruckner and Wyllie 2008). Our transcriptome results show that during fatty acid degradation pathway, the expression level of genes involved in β-oxidation in T1 or T2 were significantly elevated. The upregulation of these genes may be related to the degradation of aroma component n-Hexadecanoic acid and the formation of branched acetate or butyrate.

**qRT-PCR Validation**

To validate the reliability of genes experimentally (obtained from the assembled transcriptome of gene expression using RNA-Seq data), a total of 8 genes were selected for qRT-PCR. As shown in Fig. 6, all the expression patterns of these genes, as detected by qRT-PCR, were essentially consistent with those obtained from RNA-Seq data, which established
that the genes obtained from the assembled transcriptome are valid, and gene expression profiles from RNA-Seq data are reliable. Noticeably, the expression of the two genes of the valine, leucine, and isoleucine biosynthesis pathways, *MaBCAT3* (GSMUA-Achr3G15170-001) and *MaBCAT5* (GSMUA-Achr3G15180-001), with the extension of the post-harvesting stage, were significantly increased, and the trend is consistent with transcription group (File S2). This demonstrates that these two genes may dominantly regulate the synthesis of aroma substances in the late ripening period of bananas. In the linoleic acid metabolism pathway, the *MaTGL4* (GSMUA-Achr3G14540-001) gene proved to have a tendency to first increase and then decline with the extension of the post-harvesting stage, which is consistent with the transcriptome trend (File S2). The reason could be the early appearance of the gene expression. The aroma substance is released for a period of time after the expression level rises, and the T3 phase of the post-harvesting of the banana seems to be the final step of the post-harvesting process of the banana, until aroma substances accumulate to a certain degree, and gene expression is not significantly expressed relative to T2, but is maintained at a certain expression level, or even diminishes further.

**Conclusions**

Bananas are picked during the post-harvesting stage and they enter the edible stage after the post-harvesting process is completed. The physiological and biochemical indicators of banana fruits show tremendous variations at different post-harvesting stages, such as solid color, fruit hardness, fruit quality, fruit aroma. Among them, the change of VOCs in the post-harvesting process serves as one of the most vital indicators that dominantly affect the flavor of bananas. Therefore, it is of great theoretical and practical significance to study the changing rules and mechanisms of banana VOCs in different post-harvesting stages. In this study, firstly, the GC–MS technology was utilized to detect the changes of VOCs in various post-harvesting stages of bananas, and then was further combined with transcriptomics technology to discover three candidate genes involved in the synthesis of VOCs in different post-harvesting stages of bananas. The data deduced by this study not only provides a better understanding of the mechanism of banana VOC synthesis in different post-harvesting stages, but also lays out a basic guide to the design of sophisticated strategies to enhance...
banana post-harvesting aroma components through genetic improvement.

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**Author Contributions** Conceptualization, TD; Methodology, CH and SZ; Software, YL; Validation, HG and WH; Formal analysis, GD; Investigation, OS and FB; Resources, QY; Data curation, CL; Writing—original draft preparation, TD; Writing—review and editing, GY and TD; Visualization, FB; Supervision, TD and GY; Project administration, FB, TD, and CL; Funding acquisition, TD and GY. All authors have read and agreed to the published version of the manuscript.

**Declarations**

**Conflict of interest** The authors declare no conflict of interest. The funders had no role in the design of the study; in the collection, analyses, or interpretation of data; in the writing of the manuscript, or in the decision to publish the results.

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