CO₂ emissions from peat-draining rivers regulated by water pH

Abstract. Southeast Asian peatlands represent a globally significant carbon store that is destabilized by deforestation and the transformation into plantations, causing high carbon dioxide (CO₂) emissions from peat soils and increased leaching rates of peat carbon into rivers. While these high carbon leaching rates and consequently high DOC concentrations indicate global model studies assumed that CO₂ emissions from peat-draining rivers would be high, estimates based on field data suggest they are only moderate. In this study we offer an explanation for this phenomenon and show that carbon decomposition is hampered by the low pH in peat-draining rivers, which limits CO₂ production in and emissions from these rivers. We find an exponential pH limitation that shows good agreement with laboratory measurements from high latitude peat soils. Additionally, our results suggest that enhanced input of carbonate minerals increase CO₂ emissions from peat-draining rivers by counteracting the pH limitation. As such inputs of carbonate minerals occur due to human activities like deforestation of river catchments, liming in plantations and enhanced weathering projects, our study points out an important feedback mechanism of those practices.

1 Introduction

Rivers and streams emit high amounts of carbon dioxide (CO₂) to the atmosphere (Cole et al., 2007), but estimates of these emissions (0.6 – 1.8 PgC yr⁻¹) are highly uncertain (Aufdenkampe et al., 2011; Raymond et al., 2013). Studies agree that more than three-quarters of the global river CO₂ emissions occur in the tropics (Raymond et al., 2013; Lauerwald et al., 2015). River CO₂ emissions are controlled by the partial pressure difference between CO₂ in the atmosphere and in the river water (Raymond et al., 2012), whereby riverine CO₂ is fed by decomposition of organic matter that is leached from soils (Wit et al., 2015) and by the leaching of dissolved CO₂ from soil respiration (Abril and Borges, 2019; Lauerwald et al., 2020). Model-based studies suggest Southeast Asia as a hotspot for river CO₂ emissions Despite scarcity in river CO₂ measurements from Southeast Asia, studies suggest this region as a hotspot for river CO₂ emissions (Lauerwald et al., 2015; Raymond et al., 2013) due to the presence and degradation of carbon-rich peat soils.
More than about half of the known tropical peatlands are located in Southeast Asia (Dargie et al., 2017; Page et al., 2011), whereby 84% of these are Indonesian peatlands, mainly on the islands of Sumatra, Borneo and Irian Jaya (Page et al., 2011). Already in 2010, land use changes affected 90% of the peatlands located on Sumatra and Borneo (Miettinen and Liew, 2010) and turned them from CO$_2$ sinks to CO$_2$ sources (Hooijer et al., 2010). Enhanced decomposition in disturbed peatlands additionally increases the leaching of organic matter from soils into peat-draining rivers (Rixen et al., 2016; Moore et al., 2013). According to Regnier et al. (2013), land use changes remobilize about (1.0 ± 0.5) Pg of soil organic carbon per year of which 40% are decomposed in rivers and emitted as CO$_2$ to the atmosphere. The resulting CO$_2$ emissions of 0.4 PgC yr$^{-1}$ represent 33% of the total CO$_2$ emissions from rivers (Regnier et al., 2013).

Since peat soils are rich in carbon, concentrations of dissolved organic carbon (DOC) in peat-draining rivers are high and increase with increasing peat coverage of the river catchments (Wit et al., 2015). However, despite high leaching rates that cause DOC concentrations which can be more than four times higher than those in temperate regions (Butman and Raymond, 2011; Müller et al., 2015), measured and DOC concentrations, measurement-based field studies find that CO$_2$ fluxes from tropical peat-draining rivers in Southeast Asia (25.2 gC m$^{-2}$ yr$^{-1}$) hardly exceed those measured for rivers in temperate regions zones (18.5 gC m$^{-2}$ yr$^{-1}$, Wit et al., 2015; Butman and Raymond, 2011). Possible reasons that were suggested for these moderate emissions are short residence times of peat derived DOC in rivers due to the location of peatlands near the coast (Müller et al., 2015) as well as the recalcitrant nature of DOC (Müller et al., 2016) and the lack of oxygen (O$_2$, Wit et al., 2015) which both lower the rate of DOC decomposition. Borges et al. (2015) previously suggested a limitation of bacterial production and the resulting DOC decomposition in African peat-draining rivers as consequence of low pH based on observations at rivers in the Congo basin.

The assumption of low O$_2$ concentrations and pH as cause for moderate CO$_2$ emissions is supported by the regulating effect of these parameters on decomposition rates in peat soils, where pH and O$_2$ are the key parameters that limit the activity of the decomposition impelling enzyme phenol oxidase (Pind et al., 1994; Freeman et al., 2001). Studies agree that the limiting effect of oxygen on decomposition rates is accurately represented by the Michaelis-Menten kinetics (Fang and Moncrieff, 1999; Pereira et al., 2017). This approach assumes that decomposition rates are linearly limited for low O$_2$ concentrations but that there is no limitation for higher O$_2$ concentrations once they are sufficient to meet the decomposition demands (Keiluweit et al., 2016). Due to high rates of decomposition caused by the carbon rich environment and low rates of photosynthesis caused by low nutrient concentrations and dark water colors that limit light availability to algae, While peat-draining rivers are usually undersaturated with regard to atmospheric O$_2$ (Wit et al., 2015; Baum and Rixen, 2014). Still, their O$_2$ concentrations exceed those in peat soils due to gas exchange with the atmosphere (Müller et al., 2015; Rixen et al., 2008) and thus are assumed to limit decomposition rates less strongly than in peat soils (Pind et al., 1994). The same applies for the pH limitation, as pH in peat-draining rivers is usually higher than in peat soils (Pind et al., 1994). Other than for O$_2$ limitation, however, the form of the pH limitation is still subject to discussion. Linear (Sinsabaugh, 2010) as well as exponential (Williams et al., 2000; Kang et al., 2018) correlations have been stated in literature.
This study aims at quantifying the impact of pH and O$_2$ on the DOC decomposition in peat-draining rivers in order to explain the measured moderate CO$_2$ emissions from those rivers by the limiting effect of these parameters, and the resulting CO$_2$ emissions to the atmosphere. We analysed data from ten Southeast Asian peat-draining rivers with DOC concentrations between 200µmol L$^{-1}$ and 3,000µmol L$^{-1}$ and pH and O$_2$ concentrations ranging from 3.8 to 7.1 and from 50µmol L$^{-1}$ to 200µmol L$^{-1}$, respectively.

2 Materials and methods

2.1 Study area

Southeast Asian peatlands store 42Pg soil carbon across an area of 271,000km$^2$ (Hooijer et al., 2010). More than 97% of these peat soils are located in lowlands (Hooijer et al., 2006). The development of peatlands in Southeast Asia is favoured by its tropical climate with high precipitation rates that range between 120mm in July and 310mm in November with an annual mean of 2,700mm yr$^{-1}$ (Yatagai et al., 2020). Due to deforestation and conversion into plantations, today less than one-third of those Southeast Asian peatlands remain covered by peat swamp forests, while in 1990 it were more than three-quarters (Miettinen et al., 2016). Southeast Asian rivers mostly originate in mountain regions and cut through coastal peatlands on their way to the ocean (Fig. 1). Measurement data included in this study were obtained in river parts that flow through peat soils to capture the influence of peatlands on the carbon dynamics in the rivers.

**Figure 1.** Map of river catchments with the location of peat areas. Blue lines indicate the main rivers. Blue shaded areas outline the river basins and brown areas indicate peatlands.

The collective data were derived from four rivers on Borneo (Sarawak, Malaysia) and six rivers on Sumatra (Indonesia). The investigated rivers on Borneo are the Rajang, Simunjan, Sebuyau and Maludam and the rivers surveyed on Sumatra are
the Rokan, Kampar, Indragiri, Batang Hari, Musi and Siak (Fig. 1). We additionally include data from the Siak’s tributaries Tapung Kiri, Tapung Kanan and Mandau. River peat coverages range from 4% in the Musi catchment to 91% in the Maludam catchment, whereby the bigger rivers that originate in the uplands generally have lower peat coverages than smaller coastal rivers.

2.2 Campaigns and measurements

Data were derived from a total of 16 campaigns in Sumatra and Sarawak (Tab. A1). For the Indonesian rivers, ten measurement campaigns between 2004 and 2013 were conducted. We use published data from Baum et al. (2007) for the Mandau, Tapung Kanan and Tapung Kiri rivers, from Wit et al. (2015) for the Siak, Indragiri, Batang Hari and Musi rivers and from Rixen et al. (2016) for the Rokan and Kampar rivers. CO₂ measurements are available for the campaigns performed after 2008.

Table 1. Measured data from the investigated rivers.

| River       | peat coverage (%) | pH    | T (°C) | DOC (μmol L⁻¹) | O₂ (μmol L⁻¹) | CO₂ (μmol L⁻¹) | k₆₀₀ (cm h⁻¹) | F₃₃₂ (g C m⁻² d⁻¹) |
|-------------|-------------------|-------|--------|----------------|--------------|---------------|---------------|------------------|
| Musi        | 4.0 ± 0.1         | 6.9 ± 0.3 | 30.6 ± 0.3 | 244 ± 5       | 149 ± 43     | 128 ± 18      | 17 ± 4        | 2.8 ± 2.9        |
| Batang Hari | 5.4 ± 0.1         | 7.1 ± 0.3 | 30.0 ± 0.1 | 321 ± 4       | 163 ± 1      | 72 ± 1        | 17 ± 4        | 1.4 ± 0.4        |
| Indragiri   | 11.4 ± 0.2        | 6.3 ± 0.3 | 31.5 ± 0.1 | 692 ± 5       | 89 ± 3       | 171 ± 4       | 17 ± 4        | 3.8 ± 1.2        |
| Siak        | 25.9 ± 0.4        | 5.1 ± 0.5 | 30.0 ± 0.2 | 1,829 ± 601   | 53 ± 22      | 256 ± 21      | 17 ± 4        | 5.9 ± 2.6        |
| Kampar      | 27.8 ± 0.5        | 6.4 ± 0.4 | 29.4 ± 0.7 | 1,280 ± 44    | 98 ± 43      | n.d.          | n.d.          | n.d.             |
| Rokan       | 18.6 ± 0.3        | 6.5 ± 0.1 | 28.9 ± 1.1 | 781 ± 53      | 114 ± 22     | n.d.          | n.d.          | n.d.             |
| Mandau      | 48.1 ± 0.8        | 4.8 ± 0.7 | 30.3 ± 2.3 | 2,484 ± 669   | 63 ± 25      | n.d.          | n.d.          | n.d.             |
| Tapung Kanan| 53.4 ± 0.9        | 5.8 ± 0.7 | 30.3 ± 1.0 | 1,526 ± 169   | 86 ± 27      | n.d.          | n.d.          | n.d.             |
| Tapung Kiri | 3.9 ± 0.1         | 6.3 ± 0.5 | 30.8 ± 2.2 | 640 ± 162     | 132 ± 50     | n.d.          | n.d.          | n.d.             |
| Rajang      | 7.7 ± 0.1         | 6.7 ± 0.1 | 28.8 ± 1.2 | 169 ± 32      | 190 ± 26     | 92 ± 16       | 9 ± 1         | 1.9 ± 1.8        |
| Maludam     | 90.7 ± 1.5        | 3.8 ± 0.2 | 26.0 ± 0.5 | 4,031 ± 805   | 55 ± 36      | 281 ± 30      | 5 ± 2         | 6.5 ± 3.2        |
| Sebuyau     | 60.7 ± 1.0        | 4.2 ± 0.2 | 27.8 ± 0.6 | 3,026 ± 1,047 | 61 ± 26      | 279 ± 34      | 9 ± 5         | 6.4 ± 4.9        |
| Simunjan₁   | 42.9 ± 0.7        | 5.3 ± 0.4 | 28.2 ± 0.6 | 1,533 ± 559   | 107 ± 21     | 248 ± 54      | 11 ± 5        | 5.7 ± 4.9        |
| Simunjan₂   | 42.9 ± 0.7 ▲ 5.0 ± 0.3 *** | 27.9 ± 0.3 | 8,366 ± 1,694 | 52 ± 19 **    | ▲ 475 ± 67 *** | 11 ± 5        | 11.2 ± 6.5 *** |

Values are means of river campaigns. Data variability is given by the standard deviation of the measurements. * For the Simunjan, the March 2015 and July 2017 campaigns (Simunjan₁) were separated from the January 2016 and March 2017 campaigns (Simunjan₂) due to strong differences in the parameters. ** Due to technical problems during the Simunjan campaign in January 2016, these values are only based on one measurement campaign.

For the Malaysian rivers, measurements were performed in six campaigns between 2014 and 2017. We use data published by Müller-Dum et al. (2018) for the Rajang river and by Müller et al. (2015) for the Maludam campaigns in 2014 and 2015. Additional campaigns for this study were conducted in March 2015 at the Simunjan and Sebuyau rivers as well as in January 2016, March 2017 and July 2017 at the Simunjan, Sebuyau and Maludam rivers. Measurements of DOC, CO₂ and O₂ concentrations as well as pH, water temperatures (T) and gas exchange coefficients (k₆₀₀) for these additional campaigns were performed

4
in the same manner as during the 2014 Maludam campaign (Müller et al., 2015). However, due to technical problems, the CO$_2$, O$_2$ and pH data measured at the Simunjan river in 2016 were ignored for our analysis. Table 1 lists the averaged river parameters, including the catchments’ peat coverages and atmospheric CO$_2$ fluxes.

During the January 2016, March 2017 and July 2017 campaigns, concentrations of particulate inorganic carbon (PIC) in form of CaCO$_3$ were measured. Discrete water samples, taken from approximately 1 m below the water surface, were filtered through pre-weighed and pre-combusted glass fiber filters (0.7 µm) to sample particulate material within the water volume. To determine the particulate carbon (organic and inorganic), the samples were then catalytically combusted at 1,050°C and combustion products were measured by thermal conductivity using an Euro EA3000 Elemental Analyzer. The PIC was determined from the difference between this total particulate carbon and particulate organic carbon that was measured after addition of 1 molar hydrochloric acid in order to remove the inorganic carbon from the sample.

2.3 Additional parameters and catchment properties

Atmospheric CO$_2$ fluxes from the rivers were calculated from exchange coefficients and CO$_2$ concentrations according to

$$F_{CO_2} = k_{CO_2}(T) \cdot (CO_2 - K_{CO_2}(T) \cdot pCO_2^a),$$  

(1)

where $k_{CO_2}(T)$ was calculated from $k_{600}$ according to Wanninkhof (1992). $pCO_2^a$ is the atmospheric partial pressure of CO$_2$ ($\approx 400$μatm) and $K_{CO_2}$ describes the temperature dependent Henry coefficient for CO$_2$, which was calculated according to Weiss (1974). The atmospheric O$_2$ fluxes ($F_{O_2}$) were derived analogously with $k_{O_2}(T)$ calculated according to Wanninkhof (1992) and Henry coefficients for O$_2$ calculated according to Weiss (1970).

Catchment sizes were derived from Hydro-SHEDS (Lehner et al., 2006) at 15s resolution in WGS 1984 Web Mercator Projection. using Esri’s ArcMap 10.5. Sub-basins belonging to the catchments were identified using the HydroSHEDS 15s flow directions data set and added to the main basins. Catchment areas were then determined using WGS 1984 Web Mercator Projection.

Peat maps were downloaded from www.globalforestwatch.org for Indonesia and Malaysia. The Indonesian peatland map was published by the Ministry of Agriculture in 2012. The Malaysian peat map was made available by Wetlands International in 2004 and is based on a national inventory by the Land and Survey Department of Sarawak (1968). Both maps include peatlands in different conditions, from undisturbed peat swamp forest to disturbed peat under plantations, which is nowadays widespread in those countries. Peat coverage was determined from the areal extent of peatlands in the catchment divided by catchment size. Peat coverages derived using other peat maps are compared in Appendix B.

2.4 Quantification of the pH and O$_2$ impact on decomposition rates

The decomposition rate of DOC ($R$) is defined as molecules of CO$_2$ that are produced per available molecules of DOC during a specific time step and thus represents the proportionality factor between the CO$_2$ production rate and the DOC concentration:

$$R = \frac{\Delta CO_2}{DOC \cdot \Delta t} \Rightarrow \frac{\partial CO_2}{\partial t} = R \cdot DOC.$$  

(2)
As discussed before, \( R \) can be limited by \( O_2 \) concentrations and by \( pH \). We used an \( O_2 \) limitation factor that is based on the Michaelis-Menten equation \( (L_{O_2} = \frac{O_2}{K_m + O_2}) \) as suggested by Pereira et al. (2017). For \( pH \) limitation, we consider two approaches suggested in literature that are represented by an exponential limitation factor \( (L_{pH} = \exp(\lambda \cdot (pH - pH_0))) \) as suggested by Williams et al. (2000) and by a linear limitation factor \( (L_{pH} = \frac{pH}{pH_0}) \) as suggested by Sinsabaugh (2010). Considering the definition of \( pH \) as negative decadic logarithm of \( H^+ \) activity \( (\{ H^+ \}) \), the exponential limitation factor is equivalent to a correlation with \( \{ H^+ \}^{\frac{\lambda}{\ln(10)}} \).

For the exponential \( pH \) approach the \( CO_2 \) production rate due to DOC decomposition is given by

\[
\frac{\partial CO_2}{\partial t} = R_{max} \cdot L_{O_2} \cdot L_{pH} \cdot DOC = R_{max} \cdot \frac{O_2}{K_m + O_2} \cdot \exp(\lambda \cdot (pH - pH_0)) \cdot DOC,
\]

(3)

where \( R_{max} \) is the maximum decomposition rate, \( K_m \) is the Michaelis constant for \( O_2 \) inhibition that is also called the half saturation constant and gives the \( O_2 \) concentration at which \( O_2 \) limits decomposition by 50\% (Loucks and Beek, 2017), \( \lambda \) is the \( pH \) inhibition constant and \( pH_0 \) is a normalization constant that was set to 7.5 since this is reported to be the optimal \( pH \) for the activity of the decomposition impelling enzyme phenol oxidase (Pind et al., 1994; Kocabas et al., 2008). Equation (3) is only valid for \( pH \leq pH_0 \), as the limitation factor cannot be \( > 1 \). For higher water \( pH \), a different approach would be needed.

However, for the rivers in this study Eq. (3) is sufficient since their \( pH \) is \( < 7.5 \) (Tab. 1). When \( O_2 \) concentrations and water \( pH \) are high enough not to limit the decomposition rate, Eq. (3) simplifies to Eq. (2) with \( R = R_{max} \).

The dissolved inorganic carbon (DIC) concentrations in peat-draining rivers, as a first approximation, result from an equilibrium between \( CO_2 \) emissions and \( CO_2 \) production by decomposition. We acknowledge that this approximation assumes photosynthetic \( CO_2 \) consumption and direct \( CO_2 \) input from leaching to be negligible, which might not be the case for all rivers and we discuss the impact of these processes later on. Based on the initial approximation, Therefore, we optimized the parameters in Eq. (3) such that the production of \( CO_2 \) in the water volume beneath a specific surface area equals the atmospheric \( CO_2 \) flux through this area. The \( CO_2 \) production is calculated by multiplication of Eq. (3) with the product of river depth \( d \) and surface area \( A \) and the \( CO_2 \) emissions are calculated by multiplication of Eq. (1) with the surface area \( A \):

\[
d \cdot A \cdot R_{max} \cdot \frac{O_2}{K_m + O_2} \cdot \exp(\lambda \cdot (pH - pH_0)) \cdot DOC = A \cdot k_{CO_2}(T) \cdot (CO_2 - K_{CO_2}(T) \cdot pCO_2^a).
\]

(4)

Analogously, river \( O_2 \) concentrations result from an equilibrium between the atmospheric \( O_2 \) flux and \( O_2 \) consumption due to decomposition. During decomposition, the \( O_2 \) consumption is proportional to the \( CO_2 \) production \( (\Delta O_2 = -b \cdot \Delta CO_2) \). The proportionality factor \( b \) is usually \( < 1 \) since a fraction of the \( O_2 \) used for decomposition is taken from the oxygen content in the dissolved organic matter (Rixen et al., 2008). Thus, the equilibrium between \( O_2 \) consumption within the water volume and \( O_2 \) flux through the surface area can be written as

\[
-b \cdot d \cdot A \cdot R_{max} \cdot \frac{O_2}{K_m + O_2} \cdot \exp(\lambda \cdot (pH - pH_0)) \cdot DOC = A \cdot k_{O_2}(T) \cdot (O_2 - K_{O_2}(T) \cdot pO_2^a).
\]

(5)

In order to compare these dependencies to measured data, Eq. (4) and Eq. (5) were analytically solved for \( CO_2 \) and for \( O_2 \), respectively. The resulting equations are listed in Tab. 2. The analogously derived equations for \( CO_2 \) and \( O_2 \) that result from
the linear pH approach are listed in Tab. 3. Based on these equations, least squares optimizations were performed for the decomposition parameters \( R_{\text{max}} \), \( b \), \( K_m \) and \( \lambda \) such that \( \text{CO}_2(\text{DOC, pH, O}_2) \) and \( \text{O}_2(\text{DOC, pH}) \) are simultaneously optimized for the measured parameters of DOC, pH, \( T \), \( \text{CO}_2 \) and \( \text{O}_2 \).

**Table 2.** Equations to derive \( \text{CO}_2 \) and \( \text{O}_2 \) for the exponential pH approach.

\[
\begin{align*}
\text{CO}_2(\text{DOC, pH, O}_2) &= K_{\text{CO}_2}(T) \cdot \text{pCO}_2 + \frac{d \cdot R_{\text{max}} \cdot \text{DOC} \cdot \text{O}_2}{K_{\text{CO}_2}(T) + \text{O}_2} \exp\left(\lambda \cdot (\text{pH} - \text{pH}_0)\right) \\
\text{O}_2(\text{DOC, pH}) &= \left(\frac{b \cdot d \cdot R_{\text{max}} \cdot \text{DOC} \cdot \exp\left(\lambda \cdot (\text{pH} - \text{pH}_0)\right) + k_{\text{O}_2}(T) \cdot (K_m - K_{\text{O}_2}(T) \cdot \text{pO}_2)}{2 \cdot k_{\text{O}_2}(T)}\right)^2 + K_{\text{O}_2}(T) \cdot \text{pO}_2 - K_m - \frac{b \cdot d \cdot R_{\text{max}} \cdot \text{DOC} \cdot \exp\left(\lambda \cdot (\text{pH} - \text{pH}_0)\right) + k_{\text{O}_2}(K_m - K_{\text{O}_2}(T) \cdot \text{pO}_2)}{2 \cdot k_{\text{O}_2}(T)}
\end{align*}
\]

Equations to derive \( \text{CO}_2 \) from measured DOC, \( \text{pH} \) and \( \text{O}_2 \) as well as to derive \( \text{O}_2 \) from measured DOC and \( \text{pH} \). The parameters \( R_{\text{max}}, K_m, \lambda \) and \( b \) were derived by least squares optimization based on measured DOC, \( \text{pH} \), \( T \), \( \text{O}_2 \) and \( \text{CO}_2 \) data of the investigated rivers.

The equations in Tab. 2 and Tab. 3 depend on the river gas exchange coefficients for \( \text{CO}_2 \) \( (k_{\text{CO}_2}) \) and \( \text{O}_2 \) \( (k_{\text{O}_2}) \), which both depend on \( k_{600} \). Those exchange coefficients are poorly constrained and spatially as well as temporally extremely variable. The \( k_{600} \) we list in this study are based on a variety of techniques, including floating chamber measurements, calculations based on wind speed and catchment parameters and balance models of water parameters. Although all of those estimates remain highly uncertain, we find a fairly good agreement between \( k_{600} \) and river depths \( (d, \text{Fig. A1}) \). We therefore use a fixed ratio of \( k_{600}/d = (7.0 \pm 0.5) \cdot 10^{-6} \text{ mol L}^{-1} \text{ s}^{-1} \) for the least squares approximations.

**Table 3.** Equations to derive \( \text{CO}_2 \) and \( \text{O}_2 \) for the linear pH approach.

\[
\begin{align*}
\text{CO}_2(\text{DOC, pH, O}_2) &= K_{\text{CO}_2}(T) \cdot \text{pCO}_2 + \frac{d \cdot R_{\text{max}} \cdot \text{DOC} \cdot \text{O}_2}{K_{\text{CO}_2}(T) + \text{O}_2} \frac{\text{pH}}{\text{pH}_0} \\
\text{O}_2(\text{DOC, pH}) &= \left(\frac{b \cdot d \cdot R_{\text{max}} \cdot \text{DOC} \cdot \frac{\text{pH}}{\text{pH}_0} + k_{\text{O}_2}(T) \cdot (K_m - K_{\text{O}_2}(T) \cdot \text{pO}_2)}{2 \cdot k_{\text{O}_2}(T)}\right)^2 + K_{\text{O}_2}(T) \cdot \text{pO}_2 - K_m - \frac{b \cdot d \cdot R_{\text{max}} \cdot \text{DOC} \cdot \frac{\text{pH}}{\text{pH}_0} + k_{\text{O}_2}(K_m - K_{\text{O}_2}(T) \cdot \text{pO}_2)}{2 \cdot k_{\text{O}_2}(T)}
\end{align*}
\]

Equations to derive \( \text{CO}_2 \) from measured DOC, \( \text{pH} \) and \( \text{O}_2 \) as well as to derive \( \text{O}_2 \) from measured DOC and \( \text{pH} \). The parameters \( R_{\text{max}}, K_m, \lambda \) and \( b \) were derived by least squares optimization based on measured DOC, \( \text{pH} \), \( T \), \( \text{O}_2 \) and \( \text{CO}_2 \) data of the investigated rivers.

### 3 Results

#### 3.1 Correlation with peat coverage

The data presented in Tab. 1 yield a linear increase of river DOC concentration with peat coverage (Fig. 2a) as well as a negative linear correlation between river \( \text{pH} \) and peat coverage (Fig. 2b). The river \( \text{CO}_2 \) concentration shows a strong increase for peat coverages < 30\%. Yet, despite further increase in DOC concentrations, \( \text{CO}_2 \) concentrations in rivers with peat coverage > 30\% level off, resulting in a fairly constant \( \text{CO}_2 \) for peat coverages > 50\% (Fig. 2c). The river \( \text{O}_2 \) shows an opposite behaviour to the \( \text{CO}_2 \). \( \text{O}_2 \) concentrations initially decrease with increasing peat coverage and show a decline in the regression rate for high peat coverages, resulting in a minimum \( \text{O}_2 \) concentration of approximately 65\( \mu \text{mol L}^{-1} \) (Fig. 2d).

However, the Simunjan seems to be an exception. Although we found that generally \( \text{CO}_2 \) concentrations stagnate for high peat coverages, extremely high \( \text{CO}_2 \) concentrations were measured during two campaigns in the Simunjan river (Fig. 2). In
January 2016 and March 2017 (Simunjan\textsubscript{2}) DOC and CO\textsubscript{2} concentrations in the Simunjan river were significantly higher than in March 2015 and July 2017 (Simunjan\textsubscript{1}, Tab. 4). O\textsubscript{2} concentrations during these campaigns were lower ($\approx 50$µmol L$^{-1}$) than for the other Simunjan campaigns ($\approx 107$µmol L$^{-1}$), while the water pH of 5.0 was only slightly lower than during the other campaigns ($pH \approx 5.3$). The Simunjan campaigns with high DOC and CO\textsubscript{2} concentrations were accompanied by high concentrations of particulate carbonate (CaCO\textsubscript{3}, Tab. 4), while CaCO\textsubscript{3} concentrations in July 2017 were much lower.

![Figure 2](image)

**Figure 2.** Correlation of peat coverage (PC) with (a) DOC, (b) pH, (c) CO\textsubscript{2} & (d) O\textsubscript{2}. Each data point represents one river. Variability is indicated by the error bars, which are given by standard deviation. For the Simunjan river, the January 2016 and March 2017 campaigns (Simunjan\textsubscript{2}, see Tab. 1 and Tab. 4), indicated by red data points, were separated from the other Simunjan campaigns (Simunjan\textsubscript{1}) and excluded from the correlations due to strong deviations from the other campaigns that imply an additional process discussed in Sect. 4.3. Ordinary least squares approximations were used to calculate linear correlations with DOC and pH and exponential correlations with CO\textsubscript{2} and O\textsubscript{2}. Rivers included in a previous study investigating these correlations (Wit et al., 2015) are indicated by squares.

**Table 4.** Data measured in the four Simunjan campaigns.

| Campaign  | pH     | DOC (mmol L$^{-1}$) | CO\textsubscript{2} (µmol L$^{-1}$) | O\textsubscript{2} (µmol L$^{-1}$) | CaCO\textsubscript{3} (mg L$^{-1}$) |
|-----------|--------|---------------------|------------------------------------|----------------------------------|----------------------------------|
| Simunjan\textsubscript{1} Mar 2015 | 5.2 ± 0.3 | 1.7 ± 0.7 | 268 ± 71 | 99 ± 10 | n.d. |
| Simunjan\textsubscript{2} Jan 2016 | 4.5 ± 0.3$^*$ | 9.4 ± 1.2 | > 330$^{**}$ | 189 ± 9$^*$ | 0.52 ± 0.34 |
| Simunjan\textsubscript{2} Mar 2017 | 5.0 ± 0.3 | 7.4 ± 0.6 | 475 ± 97 | 52 ± 19 | 0.63 ± 0.64 |
| Simunjan\textsubscript{1} Jul 2017 | 5.4 ± 0.3 | 1.4 ± 0.3 | 227 ± 16 | 115 ± 14 | 0.07 ± 0.05 |

Values are means of measurements. Data variability is given by standard deviation of measurements. *Due to technical problems, the March 2017 pH, CO\textsubscript{2} and O\textsubscript{2} data need to be treated cautiously. **In March 2017 only a minimum CO\textsubscript{2} concentration could be derived.
3.2 Limitation of decomposition rates by $p\text{H}$ and $\text{O}_2$

In order to gain a better understanding of the $p\text{H}$ and $\text{O}_2$ impacts on decomposition rates, we examined correlations of $\text{CO}_2$ and $\text{O}_2$ concentrations that were calculated based on the dependencies derived from both the linear (Tab. 3) and exponential (Tab. 2) approach of $p\text{H}$ limitation with measured data. Figure 3 shows the correlation for linear $p\text{H}$ limitation. Coefficients of determination for the $\text{CO}_2$ and $\text{O}_2$ correlations result to $R^2 = 0.80$ and $R^2 = 0.87$, respectively.

![Figure 3](image)

**Figure 3.** Correlation between measured and calculated concentrations of (a) $\text{CO}_2$ and (b) $\text{O}_2$. Grey lines indicate the 1:1 line. Calculations were performed based on the equations in Tab. 3 which represent linear $p\text{H}$ limitation of decomposition rates. Each data point represents one river. Grey data points are excluded from the correlation since the data for these rivers (Kampar and Rokan) are based on less than three campaigns within the same season.

The decomposition parameters for this linear $p\text{H}$ approach, derived via least squares approximation of the equation in Tab. 3 to measured data, result to a Michaelis constant for $\text{O}_2$ limitation of $K_m = (390 \pm 509) \mu\text{mol} \ \text{L}^{-1}$, a maximum decomposition rate of $R_{\text{max}} = (10 \pm 11) \mu\text{mol} \ \text{mol}^{-1} \ \text{s}^{-1}$ and a fraction of $\text{O}_2$ consumption of $b = (90 \pm 25)\%$. These values represent $p\text{H}$ limitations in the rivers that lower decomposition rates and therewith $\text{CO}_2$ production by between 6\% in the Batang Hari and 49\% in the Maludam, while the $\text{O}_2$ limitations lower decomposition rates by between 71\% in the Batang Hari and 88\% in the Maludam and the Siak.

**Table 5.** Decomposition parameters optimized via least squares approximation for the exponential $p\text{H}$ approach.

| parameter | value    | unit                  |
|-----------|----------|-----------------------|
| $R_{\text{max}}$ | 4.0 ± 0.8 | $\mu\text{mol} \ \text{CO}_2 \ \text{mol}^{-1} \ \text{DOC} \ \text{s}^{-1}$ |
| $b$       | 81 ± 10  | \%                    |
| $K_m$     | 6 ± 26   | $\mu\text{mol} \ \text{L}^{-1}$ |
| $\lambda$ | 0.52 ± 0.10 |                       |

Data were derived for exponential $p\text{H}$ limitation of decomposition via least squares optimization of the equations in Tab. 2.
Figure 4 shows the CO$_2$ and O$_2$ correlations for exponential pH limitation of decomposition. The resulting correlation for CO$_2$ ($R^2 = 0.89$) is stronger than for the linear approach, while the O$_2$ correlation, with $R=0.85$, is slightly weaker. The decomposition parameters that were derived for the exponential pH limitation are listed in Tab. 5.

![Figure 4](image)

**Figure 4.** Correlation between measured and calculated concentrations of (a) CO$_2$ and (b) O$_2$. Grey lines indicate the 1:1 line. Calculations were performed based on the equations in Tab. 2 which represent exponential pH limitation of decomposition rates. Each data point represents one river. Grey data points are excluded from the correlation since the data for these rivers (Kampar and Rokan) are based on less than three campaigns within the same season.

The Michaelis constant for O$_2$ derived for the exponential limitation, with $K_m \approx 6 \mu mol \, L^{-1}$ is significantly smaller than the constant derived for linear pH limitation. The maximum decomposition rate ($R_{\text{max}} \approx 4 \mu mol \, mol^{-1} \, s^{-1}$) and the fraction of O$_2$ consumption ($b \approx 81\%$), while being in the same order of magnitude, are also smaller than for linear pH limitation. The exponential pH limitation factor results to $\lambda \approx 0.5$. O$_2$ limitation, resulting from these parameters, limits decomposition in the investigated rivers by $\leq 10\%$, while pH limitation ranges between 20% in the Batang Hari and 85% in the Maludam. The total limitation by O$_2$ and pH ranges between 23 and 87% (Tab. 6).

**Table 6.** pH and O$_2$ limitations calculated for individual rivers.

| River      | pH lim. (%) | O$_2$ lim. (%) | total lim. (%) | River      | pH lim. (%) | O$_2$ lim. (%) | total lim. (%) |
|------------|-------------|----------------|----------------|------------|-------------|----------------|----------------|
| Musi       | 28 ± 5      | 4 ± 1          | 31 ± 5         | Tapung Kanan | 59 ± 7      | 7 ± 2          | 62 ± 7         |
| Batang Hari| 20 ± 3      | 4 ± 1          | 23 ± 4         | Tapung Kiri | 46 ± 6      | 4 ± 1          | 49 ± 7         |
| Indragiri  | 46 ± 6      | 6 ± 2          | 50 ± 7         | Rajang     | 34 ± 5      | 3 ± 1          | 36 ± 5         |
| Siak       | 71 ± 7      | 10 ± 5         | 74 ± 8         | Maludam    | 85 ± 5      | 10 ± 4         | 87 ± 6         |
| Kampar     | 43 ± 6      | 6 ± 1          | 46 ± 7         | Sebuyau    | 83 ± 6      | 9 ± 4          | 83 ± 6         |
| Rokan      | 40 ± 6      | 5 ± 1          | 43 ± 6         | Simunjan   | 68 ± 7      | 5 ± 1          | 70 ± 7         |
| Mandau     | 76 ± 7      | 9 ± 3          | 78 ± 7         |            |             |                |                |

Fraction by which the decomposition is lowered due to the impact of pH and O$_2$, calculated based on the limitation factors in Eq. (3) and the parameters in Tab. 5 according to pH lim. = $(1 - L_{\text{pH}})$, O$_2$ lim. = $(1 - L_{O_2})$ and total lim. = $(1 - L_{\text{pH}} \cdot L_{O_2})$. 

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4 Discussion

4.1 Carbon dynamics in peat-draining rivers

The linear correlations observed between peat coverage and DOC (Fig. 2a) as well as pH (Fig. 2b) agree with results by Wit et al. (2015) and confirm the importance of peat soils as a major DOC source to these rivers, whereas the decomposition of DOC and leaching of organic acids from peat areas lower the pH. The initial increase of CO₂ concentrations (Fig. 2c) and decrease of O₂ concentrations (Fig. 2d) with peat coverage can be explained by increased DOC decomposition due to higher DOC concentrations and also agrees with the results of Wit et al. (2015).

The CO₂ stagnation we observe for rivers of higher peat coverages (Fig. 2c) agrees with moderate CO₂ emissions that were stated for those rivers (Müller et al., 2015; Moore et al., 2013) and according to Eq. (3) can be explained by the pH limitation. A similar pattern of stagnating CO₂ concentrations has been observed in river sections of high DOC at the Congo river (Borges et al., 2015). The CO₂ and DOC concentrations measured in these rivers are comparable to those measured in our study, indicating that the underlying process is valid not only for Southeast Asian rivers but for tropical peat-draining rivers in general.

4.2 Exponential pH limitation of decomposition rates

As shown, we were able to reproduce the stagnation in CO₂ and O₂ concentrations by introducing O₂ and pH limitations for decomposition rates in the rivers. Model approaches of both exponential and linear pH limitation reproduce the observed stagnation in CO₂ and O₂ concentrations and result in reasonably good correlations with the measured concentrations (Fig. 4 and Fig. 3).

The fractions of O₂ consumption by decomposition that we derived for both approaches, with \( b = (81 \pm 10)\% \) and \( b = (90 \pm 25)\% \), agree with the fraction of 0.8 that was calculated based on the oxygen to carbon ratio in peat soils (Rixen et al., 2008). The maximum decomposition rates of 4\( \mu \text{mol mol}^{-1} \text{s}^{-1} \) for the exponential approach and 10\( \mu \text{mol mol}^{-1} \text{s}^{-1} \) for the linear approach agree with global soil phenol oxidase activity data published by Sinsabaugh et al. (2008) that stated global average soil phenol oxidase activity of 70.6\( \mu \text{mol h}^{-1} \) per g organic matter. For a carbon content in organic matter of 38\( \mu \text{mol g}^{-1} \) (Sinsabaugh, 2010) this represents approximately 0.5\( \mu \text{mol mol}^{-1} \text{s}^{-1} \), while sites of high phenol oxidase activity are listed with up to 3\( \mu \text{mol mol}^{-1} \text{s}^{-1} \) (Sinsabaugh et al., 2008).

However, we assume the exponential limitation to be more realistic than the linear limitation as it is better in representing river CO₂ especially for high CO₂ concentrations which are most strongly effected by the pH limitation. This assumption is supported by the unrealistically high O₂ limitation resulting from the linear approach, which yields a Michaelis constant of \( K_m \approx 390\mu \text{mol L}^{-1} \). Since the Michaelis constant represents the O₂ concentration at which decomposition is limited by 50\% a Michaelis constant that, as in this case, is higher than the O₂ concentration in atmospheric equilibrium (\( \approx 280\mu \text{mol L}^{-1} \)) would imply an oxygen deficit at atmospheric conditions that does not exist (Vaquer-Sunyer and Duarte, 2008). In literature,
Michaelis constants between 1 and 40 µmol L\(^{-1}\) are suggested for the O\(_2\) impact on phenol oxidase, depending on the phenolic species (Fenoll et al., 2002).

The Michaelis constant for O\(_2\) derived with exponential pH limitation (\(K_m \approx 6\) µmol L\(^{-1}\)) is in good agreement with the literature data of 1 to 40 µmol L\(^{-1}\) (Fenoll et al., 2002). Its large uncertainty (> 400%, Tab. 5) is caused by relatively high concentrations of O\(_2\) in the rivers. Due to exchange with atmospheric O\(_2\) the concentrations in all rivers exceed the median O\(_2\) threshold to lethal hypoxic conditions of 50 µmol L\(^{-1}\) (Vaquer-Sunyer and Duarte, 2008). Thus, the O\(_2\) limitation in peat-draining rivers is relatively small (between 3 and 10%, Tab. 6) and consequentially a majority of the limitation is caused by the low pH in peat-draining rivers that we found to limit the decomposition rates in rivers of high peat coverage (low pH) by up to 85% (Tab. 6).

The calculated exponential pH coefficient is \(\lambda = 0.5 \pm 0.1\). Thus, in terms of H\(^+\) activity the correlation is given by \(\left\{\text{H}^+\right\}^{0.5}\), which roughly equals the fifth root of \(\left\{\text{H}^+\right\}\). The derived limitation coefficient is similar to coefficients reported for high latitude peat soils (\(\lambda = 0.65\) & \(\lambda = 0.77\)) that were determined via laboratory measurements of phenol oxidase activity (Williams et al., 2000). The fact that the exponential inhibition by pH can be found in those high latitude peat soils as well as in tropical peat-draining rivers suggests that the investigated correlations and processes are also relevant in other regions and that soil and water pH are important regulators of global carbon emissions.

As mentioned before, our results do neglect the direct leaching of CO\(_2\) from soils and the consumption of CO\(_2\) by autotrophic production within the rivers. Since CO\(_2\) leaching rates are likely higher for peat soils than for mineral soils (Kang et al., 2018) and autotrophic production is limited in peat-draining rivers (Wit et al., 2015), both of these processes would work against the observed recession in CO\(_2\) growth. This indicates that exclusion of those processes could cause underestimation of the limitation factors rather than overestimation.

4.3 Disruption of the pH limitation by carbonates

Typically, concentrations of particulate carbonate in peat-draining rivers are low (Wit et al., 2018). However we observed high CaCO\(_3\) concentrations for the Simunjan\(_2\) campaigns, which show high DOC and CO\(_2\) concentrations (Tab. 4). Possible causes for high carbonate concentrations during these campaigns could be increased erosion of mineral soils due to deforestation in mountain regions upstream or liming practices in plantations along the river. In either case, high carbonate concentrations at such a low pH indicate high dissolution of carbonates which might have counteracted a decrease in pH due to decomposition of DOC. This seems to have suspended the natural pH limitation of decomposition in peat-draining rivers which could explain the high CO\(_2\) concentrations observed during those two Simunjan campaigns (Tab. 4).

4.4 Implications and outlook

The stagnation in CO\(_2\) we observe for high peat coverages provides an explanation for the moderate CO\(_2\) emissions measured from rivers of high carbon content disagreement between model studies that state extremely high CO\(_2\) emissions from Southeast Asian rivers (Raymond et al., 2013; Lauerwald et al., 2015) and measurement-based studies that state rather moderate emission
rates (Wit et al., 2015; Müller et al., 2015). The pH limitation of decomposition that we derive to explain the observed CO₂ stagnation should be included to improve future model studies and accurately capture river CO₂ emissions from tropical peat areas.

The response on carbonate enrichment that we observe at the Simunjan river represents another important process that should be considered for anthropogenic activities like liming and enhanced weathering. Liming is a common practice to enhance soil fertility in plantations and enhanced weathering is a carbon dioxide removal strategy (Field and Mach, 2017) during which atmospheric CO₂ is transformed into carbonates (Beerling et al., 2020). The resultant increase in carbonate concentrations and pH could cause a strong increase of decomposition rates and thereby CO₂ production and emission that would counteract the CO₂ uptake, which is not included in current estimates of enhanced weathering efficiencies (Taylor et al., 2016; Beerling et al., 2020).

5 Conclusions

Our study shows that CO₂ concentrations in and emissions from Southeast Asian rivers stagnate for high peat coverages of the river catchments. Despite further increases in river DOC concentrations, CO₂ concentrations are fairly constant for peat coverages > 50%. We found that this stagnation is caused by low water pH in rivers of high peat coverage that hampers decomposition rates. This process provides an answer to the question why, in contrast to the high DOC export, CO₂ emissions from tropical peat-draining rivers are more moderate.

We found an exponential limitation of decomposition by pH. Our calculations suggest that the low pH in rivers of high peat coverage reduces decomposition rates and thereby CO₂ production within the rivers by up to 85%. Although this study is based on measurements in Southeast Asian peat-draining rivers, comparisons to laboratory studies of decomposition in temperate peat soils suggest that the investigated correlations and processes are also relevant in other regions and that soil and water pH are important regulators of global carbon emissions.

As observed in the Simunjan river, one cause for increased water pH in peat-draining rivers can be the input of carbonates. We found that CO₂ concentrations during the Simunjan campaigns that were accompanied by enhanced concentrations of suspended carbonates were significantly higher than those during campaigns of low carbonate concentrations, resulting in CO₂ emissions from this river that were increased by almost 100%. We discussed that sources for enhanced carbonate concentrations can be rock weathering or soil erosion upstream of coastal peatland areas, or liming practices in plantations along the rivers, which are common practice to improve plant growth on acidic soils.

This carbonate impact should be considered when discussing the efficiency of enhanced weathering, which is discussed as one of the possible measures to extract and bind anthropogenic CO₂ by transferring it to carbonate. The resultant pH increase, in regions of high peat coverage could lead to enhanced decomposition and thereby emissions of CO₂ from rivers and soils.

Further studies are needed to quantify the impact of the derived processes on enhanced weathering efficiencies.
Author contributions. AK performed the analysis and led the writing of the paper jointly with TR and TW. DM provided calculations of catchment parameters and in-depth comments on the manuscript. MM coordinated the field data collection in Malaysia. JN contributed to the data interpretation. All authors discussed results and commented on the manuscript.

Competing interests. The authors declare that they have no conflict of interest

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Appendix A: Additional Figures & Tables

Figure A1. Correlation between atmospheric exchange coefficients ($k_{600}$) and river depth. A linear correlation reveals a slope of $k_{600}/d = (2.5 \pm 0.2) \text{ cm h}^{-1} \text{ m}^{-1} = (7.0 \pm 0.5) \cdot 10^{-6} \text{ s}^{-1}$. 
Table A1. List of river campaigns

| River     | 03.04 | 09.04 | 08.05 | 03.06 | 04.06 | 11.06 | 03.08 | 10.09 | 10.12 | 04.13 | 03.14 | 03.15 | 01.16 | 08.16 | 03.17 | 07.17 |
|-----------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
| Maludam   | −     | −     | −     | −     | −     | −     | −     | −     | −     | −     | ✓     | ✓     | ✓     | −     | ✓     | ✓     |
| Sebuyau   | −     | −     | −     | −     | −     | −     | −     | −     | −     | −     | ✓     | ✓     | ✓     | −     | ✓     | ✓     |
| Simunjan  | −     | −     | −     | −     | −     | −     | −     | −     | −     | −     | ✓     | ✓     | ✓     | −     | ✓     | ✓     |
| Rajang    | −     | −     | −     | −     | −     | −     | −     | −     | −     | −     | ✓     | ✓     | ✓     | −     | ✓     | ✓     |
| Musi      | −     | −     | −     | −     | −     | −     | −     | −     | −     | ✓     | ✓     | ✓     | −     | −     | −     | −     |
| Batang Hari| −     | −     | −     | −     | −     | −     | −     | −     | −     | ✓     | ✓     | ✓     | −     | −     | −     | −     |
| Indragiri | −     | −     | −     | −     | −     | −     | −     | −     | −     | ✓     | ✓     | ✓     | −     | −     | −     | −     |
| Kampar    | −     | −     | −     | −     | −     | ✓     | −     | ✓     | ✓     | −     | −     | −     | −     | −     | −     | −     |
| Rokan     | −     | −     | −     | −     | −     | ✓     | −     | ✓     | ✓     | −     | −     | −     | −     | −     | −     | −     |
| Siak      | ✓     | ✓     | ✓     | ✓     | ✓     | ✓     | −     | ✓     | −     | ✓     | −     | −     | −     | −     | −     | −     |
| Mandau    | ✓     | ✓     | ✓     | ✓     | ✓     | −     | −     | −     | −     | −     | −     | −     | −     | −     | −     | −     |
| Tapung Kanan | ✓   | ✓     | ✓     | ✓     | ✓     | ✓     | −     | −     | −     | −     | −     | −     | −     | −     | −     | −     |
| Tapung Kiri | ✓   | ✓     | ✓     | ✓     | −     | −     | −     | −     | −     | −     | −     | −     | −     | −     | −     | −     |
Appendix B: Comparison of different peat coverage estimates

Different peat maps are available for Southeast Asia and the approaches to determine peat coverage of river catchments were inconsistent among different studies cited in our paper. We want to show here that the choice of a data product is crucial for the determination of peat coverage. We are comparing three different products (Tab. B1): The FAO Digital Soil Map of the World, Country products downloaded at Global Forest Watch and the Center for International Forestry Research (CIFOR) Wetlands distribution.

Table B1. Different data products used to assess peatland extent in the catchments.

| Data Product        | Product                                                                 | Coordinate System | Reference                                                                                     | Website                                                                 | Notes                                                                 |
|---------------------|-------------------------------------------------------------------------|-------------------|-----------------------------------------------------------------------------------------------|------------------------------------------------------------------------|----------------------------------------------------------------------|
| FAO                 | Food and Agriculture Organization of the United Nations (FAO): Digital Soil Map of the World | WGS 1984          | FAO Land and Water Development Division. Digital Soil Map of the World. Version 3.6. Rome, Italy 2003. | [link](http://www.fao.org/geonetwork/srv/en/metadata.show?id=14116)     | Peatlands were identified as Histosols. On Sumatra and Borneo, these are Dystric Histosols (“Od”) |
| GFW                 | Global Forest Watch Country products                                     | WGS 1984          | Indonesia: Ministry of Agriculture. Indonesia peat lands, 2012.                                | [link](www.globalforestwatch.org)                                       |                                                                      |
| CIFOR               | Center for International Forestry Research (CIFOR): Tropical and Subtropical Wetlands Distribution version 2. | WGS 1984          | Data product: Gumbricht et al. (2018); Related publication: Gumbricht et al. (2017)           | [link](https://data.cifor.org/dataset.xhtml?persistentId=doi:10.17528/CIFOR/DATA.00058) | Of the three available files, the product used was TROP_SUBTROP_PeatV21_2016_CIFOR.7z |

Those three products lead to highly different results (Tab. B2). We observed a tendency that CIFOR leads to smaller peat coverage than FAO and GFW. This is because CIFOR misses some, but not all peat areas that are known to be under industrial plantations. Gumbricht et al. (2017) already pointed out that their model underestimates peatland area in Sumatra because peats are largely drained, which the model does not capture. However, in the Musi and Batang Hari catchment, CIFOR sees larger peat areas than FAO and GFW, which means that some peatlands might be missing in those maps.
We decided to use the GFW maps for several reasons: 1) CIFOR seems to miss peat under industrial plantations, which is still relevant for river carbon dynamics. Therefore, we chose not to use the CIFOR maps. 2) Between GFW and FAO, GFW is more recent than FAO for Indonesia. For Sarawak (Malaysia), both are based on the 1968 soil map by the Land Survey Department, but FAO uses a 10-fold coarser scale than the 1968 soil map (1:5,000,000 compared to 1:500,000). Thus, the GFW product was used. & 3) GFW maps are based on official information, and we believe that the local authorities would know best about the peatland distribution in their country.

Similar to the peat coverage, the publications from which we use data in our study all had different approaches to determining catchment size – either including (Müller-Dum et al., 2018) or excluding (Wit et al., 2015) smaller sub-catchments. In our study, we aimed to unify those different approaches. Therefore, we recalculated catchment areas from one single data product (HydroSHEDS, (Lehner et al., 2006)) including sub-catchments that were identified using HydroSHEDS flow directions. The Simunjan catchment is included in the bigger Sadong catchment in HydroSHEDS. Therefore, it was manually delineated using HydroSHEDS flow directions.

| River name | Catchment (km²) | PC GFW | PC CIFOR | PC FAO |
|------------|----------------|--------|----------|--------|
| Batang Hari| 43,778         | 5.4 ± 0.1 | 6.8 ± 0.1 | 5.0 ± 0.1 |
| Indragiri  | 17,713         | 11.4 ± 0.2 | 9.6 ± 0.1 | 8.6 ± 0.1 |
| Kampar     | 23,610         | 27.8 ± 0.4 | 20.2 ± 0.2 | 18.8 ± 0.3 |
| Musi       | 57,602         | 4.0 ± 0.1 | 11.3 ± 0.1 | 3.7 ± 0.1 |
| Rokan      | 19,953         | 18.4 ± 0.3 | 8.8 ± 0.1 | 30.3 ± 0.5 |
| Siak       | 11,719         | 25.9 ± 0.4 | 14.8 ± 0.1 | 27.2 ± 0.4 |
| Maludam    | 91             | 90.7 ± 1.4 | 82.3 ± 1.1 | 100.0 ± 1.5 |
| Rajang     | 51,699         | 7.7 ± 0.1 | 7.4 ± 0.1 | 10.6 ± 0.2 |
| Sebuyau    | 451            | 60.7 ± 0.9 | 41.2 ± 0.4 | 75.8 ± 1.2 |
| Simunjan   | 755            | 42.9 ± 0.7 | 20.3 ± 0.2 | 25.9 ± 0.4 |
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