**Pleurotus ostreatus** as a Biocatalyst to Obtain Bicyclic Hydroxylactones with Three or Four Methyl Groups

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**Abstract:** The purpose of the study was to explore microbial transformations in cultures of basidiomycetes *Pleurotus ostreatus*, two bicyclic unsaturated lactones occurring in the form of two diastereoisomers. Some of these strains were able to transform unsaturated lactones into four known and three new derivatives. The structures of all lactones were established on the basis of spectroscopic data. Both substrates and products caused a complete inhibition of growth of *A. alternata* and *F. linii* strains.

**Keywords:** unsaturated lactones; hydroxylation; epoxidation; biotransformation

1. Introduction

Mushrooms of the *Pleurotus* genus are the main decomponent of plant residues. They grow mainly on dead wood, which contains lignocellulose compounds needed for their growth. They can also parasitize on living trees, infecting them with white rot [1]. These fungi are applicable in our diet as a source of valuable low fat nutrients. Their fruiting bodies are rich in easily absorbed proteins, carbohydrates, amino acids, B vitamins (thiamine, riboflavin, and niacin), and mineral salts (calcium, phosphorus, and iron) [2–4]. Depending on whether these fungi grow wild or under greenhouse conditions, quantitative proportions of nutrients may vary, due to the nature of the substrate used for the growth of mycelium [2,3,5]. Mushrooms of the *Pleurotus* genus have found various applications. They are able to accumulate heavy metals, such as lead or copper [6–8], and biodegrade harmful compounds accumulating in the natural environment, such as aflatoxins [9], herbicides [10,11], and plasticizers [12–14]. They are also applicable in biotransformations, where they are mainly used for hydroxylation of natural terpenes and their derivatives. The most frequently observed reaction is the hydroxylation in the allylic position and subsequent oxidation of the alcohol formed [15–18].

Hydroxylactones are compounds often found in nature, especially in plants and marine organisms. These compounds are characterized by a broad spectrum of biological activity, e.g., antibacterial [19,20], cytotoxic [20,21], anti-cancer [22,23], anti-inflammatory [24–26], and antifungal [27]. Isolation of hydroxylactones from natural sources is often complicated and allows the desired compounds to be obtained in small amounts. Biotransformation is an alternative method for their acquisition in which substrates obtained by chemical synthesis are used, e.g., unsaturated lactones.
During the biotransformation of unsaturated lactones that we have carried out so far, filamentous fungi have been used. As a result, both hydroxylactones and epoxylactones were obtained [28–31]. This time, we decided to use edible mushrooms as biocatalysts. Considering the described ability of Pleurotus fungi to transform terpenes, our aim was to obtain new lactones with potential biological activity.

2. Results

Two known unsaturated lactones 1 and 2 (as mixtures of two diastereoisomers named A and B), whose structures are shown in Figure 1, were used for the biotransformation. The whole cells of seven strains of oyster mushroom Pleurotus ostreatus (Jacq.) P. Kumm were used as biocatalyst.

![Figure 1. The unsaturated lactones 1 and 2.](image)

During previous studies [31], biotransformations of lactones 1 and 2 (also as mixtures of diastereoisomers A and B) were carried out using filamentous fungi of the genus Fusarium, Penicillium, Absidia, and Syncephalastrum. As a result, six hydroxylactones and two epoxylactones were obtained. In the current studies, Pleurotus ostreatus strains were used as biocatalysts. These fungi belong to the oyster family, which are, as already mentioned before, known for their ability to hydroxylate terpenoid compounds [15–18]. Seven strains of Pleurotus ostreatus were tested during the screening biotransformation. The purpose of this step was to check whether these strains were capable of conducting the transformation of selected lactones. An attempt was also made to answer the question of whether the products obtained as a result of the biotransformations carried out with the fungi of the Pleurotus ostreatus species would be identical or different to those obtained earlier with the use of filamentous fungi [31]. In the first version, biotransformations were carried out analogously to those carried on filamentous fungi, i.e., taking samples after 7 and 14 days. It turned out, however, that in the case of fungi of the Pleurotus genus, a complete conversion was observed after one week. As a result, the procedure was changed and subsequent trials were collected after 3, 5, and 7 days. The results obtained during the screening tests are provided in the Tables 1 and 2.

| Strain   | Time (Days) | Lactone 1 A + B (%) | Lactone 3 (%) | Lactone 4 (%) |
|----------|-------------|---------------------|---------------|---------------|
| PB63S    | 3           | 27.2 + 51.0         | 8.0           | 13.8          |
|          | 5           | 3.2 + 17.6          | 30.0          | 49.2          |
|          | 7           | 0 + 8.1             | 31.8          | 60.1          |
| PUSAS    | 3           | 25.8 + 49.2         | 6.4           | 18.6          |
|          | 5           | 20.8 + 41.2         | 15.9          | 21.0          |
|          | 7           | 14.1 + 38.2         | 24.1          | 23.5          |
|          | 5           | 32.0 + 51.6         | 5.6           | 10.8          |
|          | 7           | 30.3 + 35.2         | 9.5           | 25.0          |

Table 1. Results of screening biotransformation of lactone 1 after 3, 5, and 7 days of incubation (in % according to GC).
The results presented in Table 1 demonstrate that among seven tested strains, three were able to convert the unsaturated lactone 1. Strains PBo6S, PB63S, and PUSAS transformed lactone 1 into two products: hydroxylactone 3 and epoxy-lactone 4 (Scheme 1).

The results in Table 2 show that the unsaturated lactone 2 was transformed into four different products. The PBo6S and PB63S strains converted this compound to hydroxylactone 5 and epoxy-hydroxylactone 7. Additionally, when the PBo6S strain was used for biotransformation, the formation of epoxy-lactone 6 was observed. Two successive strains, i.e., PUSAS and PB7′96, transformed the substrate mainly to the already mentioned epoxy-hydroxylactone 7 and epoxy-hydroxylactone 8 (Scheme 2).
The unsaturated lactone 1 was subjected to a biotransformation with the PB06S strain. It resulted in hydroxylactone 3, which was formed exclusively from the A-isomer, and epoxylactone 4 formed as a mixture of two isomers A and B. As a result of the chromatographic separation of the products, the pure B epoxylactone 4 isomer was partially isolated.

**Table 3.** Results of the preparative scale biotransformation of lactone 1 after 7 days of incubation (in % according to GC).

| Strain | Lactone 1 A + B (%) | Lactone 3 (%) | Isolated Yield (g/%) | Lactone 4 (%) | Isolated Yield (g/%) |
|--------|---------------------|---------------|----------------------|---------------|----------------------|
| PB06S  | 0 + 8.9             | 41.0          | 0.0254/23.1          | 50.1          | 0.0117/10.6          |

The unsaturated lactone 2 was subjected to preparative biotransformations using four strains. As a result, four derivatives were obtained: Hydroxylactone 5, epoxylactone 6, and two epoxy-hydroxylactones 7 and 8. Lactones 5 and 7 were formed from the isomer A, whereas the lactones 6 and 8 were from the B-isomer of the substrate.

In the next step, two parameters—optical purity and enantiospecificity—of lactones 3–6 and 8 were determined. These lactones were obtained as single isomers during preparative biotransformation; the results of this step are included in Table 5.

The results presented in Table 5 prove that the lactones of the three methyl groups, which are compounds 3 and 4, were characterized by high (above 80%) enantiomeric excess. An additional
methyl group significantly reduced the enantiomeric excess of products 5, 6, and 8. The best results of 35.4% and 30.3% were obtained with epoxylactone 8. All lactones were formed predominantly by the (-)-enantiomer. (Supplementary Materials, Figures S39–S43).

As mentioned in the introduction, hydroxylactones are often characterized by a variety of biological activities. Therefore, all biotransformation products were tested for antimicrobial activity against strains that are pathogens for plants or animals. There were Escherichia coli C1, Staphylococcus aureus, Candida albicans KL-1, Alternaria alternata, Fusarium linii A3, and Aspergillus niger XP. In order to compare the biological activity of substrates and biotransformation products, unsaturated lactones were also tested. The obtained results were presented as the mean ± standard deviation (SD) of optical density ∆OD values and are shown in Tables 6 and 7. (Supplementary Materials, Figures S44–S49).

**Table 5.** The values of enantiospecificity and optical purity of hydroxylactones 3–6 and 8.

| Strain     | Lactone | ee (%) | [α]D 20° |
|------------|---------|--------|----------|
| PBo6S      | 3       | 82.1   | -13.263 (C = 0.17, CHCl₃) |
|            | 4       | 86.7   | -13.423 (C = 0.20, CHCl₃) |
|            | 5       | 22.6   | -5.725 (C = 0.20, CHCl₃) |
|            | 6       | 12.4   | -3.385 (C = 0.29, CHCl₃) |
| PB63S      | 5       | 23.4   | -5.836 (C = 0.29, CHCl₃) |
| PUSAS      | 6       | 35.4   | -10.839 (C = 0.74, CHCl₃) |
| PB796      | 8       | 30.3   | -6.090 (C = 0.42, CHCl₃) |

It can be noticed that the most active compound was hydroxylactone 3. This compound completely inhibited the growth of four of the six tested strains, which included E. coli and S. aureus bacteria strains as well as filamentous fungi A. alternata and F. linii strains. In case of the A. niger strain, a three-fold increase in the lag-phase duration occurred. The unsaturated lactone 1 completely inhibited only the growth of strains of filamentous fungi. A similar effect was shown by epoxylactone 4 in relation to A. alternata and F. linii strains.

**Table 6.** Antimicrobial activity of lactone 1 and its derivatives (∆OD as the mean ± SD).

| Strain  | Control | Lactone 1 | Lactone 3 | Lactone 4 |
|---------|---------|-----------|-----------|-----------|
|         | Lag-Phase [h] | ∆OD | Lag-Phase [h] | ∆OD | Lag-Phase [h] | ∆OD | Lag-Phase [h] | ∆OD |
| E. coli | 4.0 | 1.59 ± 0.03 | 9.5 | 0.28 ± 0.01 | N/A | 0 | 9 | 0.15 ± 0.04 |
| S. aureus | 2.5 | 1.70 ± 0.05 | 5.0 | 0.56 ± 0.02 | N/A | 0 | 4.5 | 0.50 ± 0.03 |
| C. albicans | 3.0 | 1.53 ± 0.05 | 5.0 | 0.52 ± 0.04 | 1.5 | 0.77 ± 0.01 | 1.5 | 0.91 ± 0.04 |
| A. alternata | 14.5 | 1.84 ± 0.04 | N/A | 0 | N/A | 0 | N/A | 0 |
| F. linii | 11.5 | 1.91 ± 0.03 | N/A | 0 | N/A | 0 | N/A | 0 |
| A. niger | 12.5 | 2.01 ± 0.05 | N/A | 0 | 36.0 | 1.44 ± 0.06 | 32 | 0.94 ± 0.07 |

* N/A - not available.

**Table 7.** Antimicrobial activity of lactone 2 and its derivatives (∆OD as the mean ± SD).

| Strain  | Control | Lactone 2 | Lactone 5 | Lactone 6 | Lactone 7 | Lactone 8 |
|---------|---------|-----------|-----------|-----------|-----------|-----------|
|         | Lag-Phase [h] | ∆OD | Lag-Phase [h] | ∆OD | Lag-Phase [h] | ∆OD | Lag-Phase [h] | ∆OD | Lag-Phase [h] | ∆OD |
| E. coli | 4.0 | 1.59 ± 0.03 | 11 | 0.11 ± 0.03 | 9.5 | 0.18 ± 0.03 | 12.5 | 0.25 ± 0.05 | 5 | 0.18 ± 0.05 | 6.5 | 0.28 ± 0.03 |
| S. aureus | 2.5 | 1.70 ± 0.05 | 25.5 | 0.20 | 7.0 | 0.44 ± 0.03 | 3.0 | 0.56 ± 0.05 | 6.5 | 0.20 ± 0.04 | 5.5 | 0.50 ± 0.03 |
| C. albicans | 3.0 | 1.53 ± 0.05 | 5.5 | 0.40 | 2.5 | 0.61 ± 0.07 | 1.0 | 0.91 ± 0.01 | 1.5 | 0.75 ± 0.02 | 1.0 | 0.92 ± 0.03 |
| A. alternata | 14.5 | 1.84 ± 0.04 | N/A | 0 | N/A | 0 | N/A | 0 | N/A | 0 |
| F. linii | 11.5 | 1.91 ± 0.03 | N/A | 0 | N/A | 0 | N/A | 0 | N/A | 0 |
| A. niger | 12.5 | 2.01 ± 0.05 | 22.5 | 0.93 ± 0.04 | 42.5 | 0.67 ± 0.03 | 25.5 | 1.21 ± 0.06 | 23.5 | 1.24 ± 0.02 | 27.5 | 1.75 ± 0.03 |
An analysis of the results indicates that the most sensitive to the tested compounds strains were filamentous fungi *A. alternata* and *F. linii*. The unsaturated lactone 2 as well as all its derivatives, lactones 5–8, completely inhibited the growth of these microorganisms. In the presence of the *A. niger* strain, the lag-phase duration increased significantly. A similar phenomenon was also observed in case of bacterial strains. Lag-phase elongation was caused by lactones 2, 5, and 6 in case of *E. coli*, and lactones 2, 5, and 7 in case of the *S. aureus* strain.

In the next stage, the values of Minimal Inhibitory Concentration (MIC) for compounds that completely inhibit the growth of tested strains was determined. Lactones 1, 3, and 4 were tested for *E. coli, S. aureus, A. alternata*, and *F. linii* strains, and lactones 2 and 5–8—for *A. alternata* and *F. linii* strains. The study was conducted for concentrations of 0.0075 to 0.1 mg/mL. The obtained results are presented in Table 8.

| Strain      | Lactone 1 | Lactone 2 | Lactone 3 | Lactone 4 | Lactone 5 | Lactone 6 | Lactone 7 | Lactone 8 |
|-------------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|
| *E. coli*   | nt        | nt        | 0.1       | nt        | nt        | nt        | nt        | nt        |
| *S. aureus* | nt        | nt        | 0.1       | nt        | nt        | nt        | nt        | nt        |
| *A. alternata* | 0.05  | 0.05      | 0.05      | 0.05      | 0.05      | 0.1       | 0.1       | 0.1       |
| *F. linii*  | 0.05      | 0.05      | 0.05      | 0.05      | 0.025     | 0.05      | 0.1       | 0.05      |
| *A. niger*  | 0.1       | nt        | nt        | nt        | nt        | nt        | nt        | nt        |

nt—not tested.

The best result was noted for lactone 5, which completely inhibited the growth of the *F. linii* strain at 0.025 mg/mL. Lactones 1–5 were able to inhibit the growth of *A. alternata* strains at a concentration of 0.05 mg/mL. Lactones 1–4, 6, and 8 inhibited the growth of the *F. linii* strain also at a concentration of 0.05 mg/mL.

3. Discussion

Two known unsaturated lactones 1 and 2 were used for the tests. Each of these compounds formed as a mixture of two diastereoisomers (A and B) (Supplementary Materials, Figures S1–S4). In case of compound 1, the difference concerned the spatial orientation of the methyl group at the C-7 carbon. This group was located in the same plane as the lactone ring (1A) or across this plane (1B). In case of compound 2, larger differences were observed. For the 2A-isomer, the C-O bond of the lactone ring was in the axial position, while the CH3-11 group was located across the plane of the lactone ring. In case of the 2B isomer, the C-O bond of the lactone ring was in the equatorial position, while the CH3-11 group was located in the plane of the lactone ring (Figure 2).

![Figure 2. Spatial structures of stereoisomers (A) and (B) of unsaturated lactones 1 and 2.](image-url)

As a result of the biotransformations, a number of compounds were obtained: Two hydroxylactones 2 and 5, two epoxylactones 3 and 6, and two epoxy-hydroxy lactones 7 and 8.
Analysis of the $^1$H NMR spectra showed that hydroxylactone 3 and epoxy-hydroxylactones 7 and 8 are novel compounds. The remaining products were identical to those obtained earlier as a result of biotransformation carried out with filamentous fungi. Hydroxylactone 4 was obtained from substrate 1 using the Absidia cylindrospora AM336 strain. Hydroxylactone 5 was obtained from substrate 2 using Syncphalastrum racemosum AM105 and Absidia cylindrospora AM336 strains, while epoxylactone 6 was from the Penicillium vermiculatum AM30 strain [31]. It is worth noting that as a result of the biotransformations carried out using filamentous fungi, hydroxylactone 5 was obtained as a mixture of both isomers A and B. However, by using fungi of the Pleurotus genus, the lactone 5 was obtained only from the A-isomer. Since the structure of the known biotransformation products have been described previously [31], here, a detailed analysis of the $^1$H NMR spectra of the new compounds (3, 7, and 8) is presented.

In case of substrate 1, hydroxylactone 3 was the new product. This compound was formed by hydroxylation of the C-7 tertiary carbon, as evidenced by the lack of signal from the proton H-7. A comparison of the $^1$H NMR spectra of both the substrate A and B isomers, in particular the chemical shifts of protons H-1, H-6, and the group, CH$_3$-11, allowed the statement that lactone 3 formed by hydroxylation of the isomer A.

As a result of the biotransformation of substrate 2, two new compounds—epoxy-hydroxylactones 7 and 8—were obtained. By analyzing the results obtained from the course of the screening biotransformation, it was found that compound 7 formed as a result of epoxidation of hydroxylactone 3, and compound 8 by hydroxylation of epoxylactone 6. This statement was also confirmed by analysis of the $^1$H NMR spectra of biotransformation products. In case of lactones 5 and 7, the differences occurring in the chemical shifts of protons H-1, H-2, H-3, and the CH$_2$-5 group can be noticed. This change is mainly affected by the chemical protons H-2 and H-3, for which the mentioned values changed from 5.76 and 5.86 ppm (5) to 2.96 and 3.58 ppm (7), respectively. This change indicates epoxidation of the double bond in the existing hydroxylactone. Epoxidation is also confirmed by the observed shift of the signal coming from the H-1 proton from 4.62 (5) to 4.77 ppm (7). However, when considering the spectroscopic data on compound 8, it can be seen that its spectrum is very similar to the spectrum of epoxylactone 6. The only difference relates to the shift of the signal derived from the methyl group, CH$_3$-10, from 1.10 (6) to 3.48 ppm (8) and the change in the multiplicity of this signal from the singlet to the doublet of the doublet. This means that in the already existing epoxylactone, the CH$_3$-10 group was hydroxylated.

Biotransformations of the unsaturated compounds described in the literature using fungi of the Pleurotus genus are mainly based on hydroxylation in the allylic position [15–18]. In our case, this type of reaction was not observed. Products were formed by hydroxylation of a tertiary carbon or double bond epoxidation. Such reactions are rather characteristic of filamentous fungi [28–31]. The formation of epoxy-hydroxylactones was also very interesting. Such compounds have not been observed by us so far. Their formation indicates that Pleurotus ostreatus are characterized by a completely different metabolism than the previously used filamentous fungi.

Studies carried out to determine the antimicrobial activity of substrates and biotransformation products gave very interesting results. The unsaturated lactone 1 completely inhibited the growth of tested strains of filamentous fungi (A. alternata, F. linii, A. niger). In case of bacterial and yeast strains, partial inhibition of growth and an approximately two-fold increase in the lag-phase duration was observed. The introduction of a hydroxy group to the lactone ring resulted in an increase in the ability to inhibit the growth of microorganisms by compound 3. Four of the six tested strains were found to be sensitive to this lactone. In turn, epoxidation of the double bond negatively influenced the degree of inhibition of the growth of the studied microorganisms.

In case of unsaturated lactone 2 and its derivatives, no relationship was observed between the structure of the compound and its ability to inhibit or limit the growth of the tested microorganisms. All compounds were able to completely inhibit the growth of only two strains: A. alternata and F. linii strains. In other tested strains, partial inhibition of growth and prolongation of the lag-phase duration was observed.
Compounds capable of completely inhibiting the growth of microorganisms were tested for determination of the MIC. The most effective of these was hydroxylactone 5, already active at 0.025 mg/mL, against the F. linii strain. Of the remaining lactones, the majority showed activity at a concentration of 0.05 mg/mL against A. alternata (compounds 1–5) and F. linii (compounds 1–4, 6, 8) strains.

4. Materials and Methods

4.1. General Methods

The products obtained during biotransformation were purified using preparative column chromatography. Silica gel (Kieselgel 60, 230–400 mesh, Merck, Darmstadt, Germany) was used as the adsorbent. The mixture of hexane and acetone in a volume ratio of 3:1 was used as an eluent. The purity of all compounds and progress of biotransformation was checked by gas chromatography GC analysis carried out on an Agilent Technologies 6890N instrument (Agilent Technologies, Santa Clara, CA, USA) using a DB-5HT column (cross-linked methyl silicone gum, 30 m × 0.32 mm × 0.25 μm). The conditions were as follows: Injector 150 °C, detector (FID) 300 °C, column temperature 100 °C, ramp 100–240 °C at a rate of 25 °C/min, ramp 240–300 °C at a rate of 40 °C/min. The enantiomeric excesses of the products of biotransformation were determined by Agilent Technologies 6890N GC analysis using the chiral column CP-Chirasil-DEX B-325 (25 m × 0.25 mm × 0.25 μm, Supelco, Bellefonte, PA, USA), injector 200 °C, detector (FID) 200 °C, initial column temperature 130 °C, ramp 130–160 °C at a rate of 2 °C/min, ramp 160–200 °C at a rate of 10 °C/min and hold 1 min at 200 °C (for compounds 4, 6, 8). For compounds 5 and 9, enantiomeric excesses were determined by a Varian CP3380 instrument using the chiral column Gamma DEX 325 (30 m × 0.25 mm × 0.25 μm, Supelco, Bellefonte, PA, USA), injector 150 °C, detector (FID) 200 °C, initial column temperature 130 °C, ramp 130–160 °C at a rate of 0.7 °C/min, ramp and 160–200 °C at a rate of 20 °C/min. High resolution mass spectrometry analysis using a Waters LCT Premier XE instrument (ESI ionization, Waters, Milford, MA, USA) was used for confirmation of the molar masses of the obtained compounds. 1H NMR and 13C NMR spectra were recorded in a CDCl3 solution on a JEOL DeltaTM 400 MHz spectrometer (JEOL USA, Inc., Peabody, MA, USA). Jasco P-2000 polarimeter (Jasco, Easton, PA, USA) was used to determined optical rotations. Tested compounds were dissolved in chloroform, and their concentration was denoted in g/100 mL. Boetius apparatus was used to determine the melting points.

4.2. Organic Synthesis

The NMR spectra of two known unsaturated lactones 1 and 2 were obtained according to the procedure described in our earlier article [31]. Their NMR spectra are described below:

4,4,7-trimethyl-9-oxabicyclo[4.3.0]non-2-en-8-one (1). 1H NMR (400 MHz, CDCl3): 5.86–5.91 (m, 2H, H-3A, H-3B), 5.74–5.80 (m, 2H, H-2A, H-2B), 4.78 (dd, J = 5.2 and 4.8 Hz, 1H, H-1A), 4.59 (t, J = 4.8 Hz, 1H, H-1B), 2.93 (quintet, J = 7.6 Hz, 1H, H-7B), 2.60–2.66 (m, 1H, H-6B), 2.41 (q, J = 7.6 Hz, 1H, H-7A), 2.44–2.30 (m, 1H, H-6A), 1.52 (dd, J = 13.6 and 5.2 Hz, 1H, one of CH3-5A), 1.42 (dd, J = 13.2 and 4.4 Hz, 1H, one of CH3-5B), 1.35 (d, J = 7.6 Hz, 3H, CH3-11A), 1.20–1.33 (m, 1H, one of CH3-5A), 1.18 (d, J = 7.2 Hz, 3H, CH3-11B), 1.04–1.11 (m, 1H, one of CH3-5B), 1.04 (s, 3H, CH3-10B), 1.03 (s, 3H, CH3-10A), 0.99 (s, 3H, CH3-9B), 0.98 (s, 3H, CH3-9A). 13C NMR (100 MHz, CDCl3): 179.82 (C-8B), 178.91 (C-8A), 145.38 (C-3B), 144.89 (C-3A), 119.91 (C-2A), 119.76 (C-2B), 73.30 (C-1A), 73.20 (C-1B), 43.24 (C-7A), 40.12 (C-7B), 38.65 (C-6A), 38.48 (C-6B), 35.35 (C-5A), 33.59 (C-5B), 32.00 (C-4B), 31.85 (C-4A), 30.56 (C-10B), 30.21 (C-10A), 27.28 (C-9A), 26.83 (C-9B), 15.61 (C-11A), 9.22 (C-11B).

4,4,6,7-trimethyl-9-oxabicyclo[4.3.0]non-2-en-8-one (2). 1H NMR (400 MHz, CDCl3): 5.87 (d, J = 10.0 Hz, 1H, H-3B), 5.75 (dd, J = 10.0 and 5.2 Hz, 1H, H-2B), 5.66 (d, J = 10.4 Hz, 1H, H-3A), 5.53 (dd, J = 10.4 and 2.8 Hz, 1H, H-2A), 4.47 (s, 1H, H-1A), 4.25 (d, J = 5.2 Hz, 1H, H-1B), 2.75 (q, J = 7.2 Hz, 1H, H-7A), 2.41 (q, J = 7.2 Hz, 1H, H-7B), 1.67 (d, J = 15.2, 1H, one of CH3-5A), 1.67 (d, J = 15.2 Hz, 1H, one of CH3-5A), 1.44 (d, J = 15.2 Hz, 1H, one of CH3-5A), 1.23 (d, J = 14.0 Hz, 2H, CH2-5B), 1.21 (s, 3H,
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CH$_3$-12B), 1.11 (d, $J = 7.2$ Hz, 3H, CH$_3$-11B), 1.09 (s, 3H, CH$_3$-10B), 1.07 (d, $J = 6.4$ Hz, 3H, CH$_3$-11A), 1.07 (s, 3H, CH$_3$-12A), 1.02 (s, 3H, CH$_3$-10A), 1.02 (s, 3H, CH$_3$-9B), 0.99 (s, 3H, CH$_3$-9A), $^{13}$C NMR (100 MHz, CDCl$_3$): 179.00 (C-8A), 178.59 (C-8B), 144.75 (C-3B), 140.65 (C-3A), 120.67 (C-2A), 117.64 (C-2B), 81.03 (C-1A), 78.41 (C-1B), 49.98 (C-5A), 41.16 (C-7A), 40.51 (C-4A), 40.31 (C-4B), 38.39 (C-5B), 33.17 (C-10A), 33.05 (C-9B), 31.85 (C-6B), 31.59 (C-6A), 29.52 (C-12A), 28.51 (C-10B), 23.13 (C-12B), 22.95 (C-9A), 7.70 (C-11A), 7.31 (C-11B).

### 4.3. Biotransformation

#### 4.3.1. Microorganisms

The whole cells of seven strains of oyster mushroom (*Pleurotus ostreatus* (Jacq.) P. Kumm) were obtained from the collection of Department of Fruit, Vegetable and Mushroom Technology at the University of Life Sciences in Lublin. The isolates were from Poland (PB63S, PBo6S, PxS, P112), Japan (PB8, PB7’96), and the USA (PUSAS). All strains are available in the collection stored in Department of Plant Protection, Wroclaw University of Environmental and Life Sciences. All these strains were cultivated on PDA (Potato Dextrose Agar) medium and stored in refrigerator at 4°C.

#### 4.3.2. Screening Procedure

Tested strains were grown in 300 mL Erlenmeyer flasks containing 100 mL of medium composed of 3% glucose and 1% peptone dissolved in distilled water. After 5 days, 10 mg of the substrate dissolved in 1 mL of acetone was added to the growing mycelium. After 3, 5, and 7 days, one third of the contents of the flask was removed, and the whole was extracted with methylene chloride (15 mL) and analyzed on GC (DB-17 column).

#### 4.3.3. Preparative Biotransformation

Preparative biotransformations were carried out analogously to screening biotransformations, the difference being that 10 Erlenmeyer flasks with a capacity of 300 mL were used. To each flask with growing mycelium, 10 mg of substrate was added (total of 100 mg). After 7 days, the whole was extracted with methylene chloride (3 × 40 mL). The combined organic fractions were dried over MgSO$_4$, and then the solvent was evaporated. The obtained products were purified on a chromatography column (silica gel, hexane: acetone 3:1). As a result of the biotransformation carried out, six products were obtained, whose physicochemical data (for new compounds) and spectral (for all compounds) are listed below.

**7-Hydroxy-4,4,7-trimethyl-9-oxabicyclo[4.3.0]non-2-en-8-one (3).** White solid, m.p. = 72–73°C, $^1$H NMR (400 MHz, CDCl$_3$): 5.94 (d, $J = 10.0$ Hz, 1H, H-3), 5.83 (dd, $J = 10.0$ and 4.8 Hz, 1H, H-2), 5.00 (dd, $J = 4.8$ and 4.0 Hz, 1H, H-1), 2.47 (dt, $J = 14.4$ and 4.8 Hz, 1H, H-6), 1.41 (dd, $J = 14.4$ and 4.8 Hz, 1H, one of CH$_2$-5), 1.41 (s, 3H, CH$_3$-11), 1.05 (s, 3H, CH$_3$-10), 1.00 (s, 3H, CH$_3$-9), 0.86 (m, 1H, one of CH$_2$-5), $^{13}$C NMR (100 MHz, CDCl$_3$): 177.14 (C-8), 145.21 (C-3), 119.63 (C-2), 34.05 (C-7), 73.01 (C-1), 42.07 (C-6), 34.05 (C-5), 32.13 (C-4), 30.38 (C-10), 26.80 (C-9), 19.27 (C-11), ESIHRMS: calcd for C$_{11}$H$_{16}$O$_3$Na, m/z 219.0997 (M + H)$^+$, found 219.1004. (Supplementary Materials, Figures S5–S10).

**2,3-Epoxy-4,4,7-trimethyl-9-oxabicyclo[4.3.0]nonan-8-one (4).** $^1$H NMR (400 MHz, CDCl$_3$): 4.78 (d, $J = 5.6$ Hz, 1H, H-1A), 4.63 (dd, $J = 4.4$ and 1.2 Hz, 1H, H-1B), 3.52 (dd, $J = 3.6$ and 1.2 Hz, 1H, H-2B), 3.42 (dd, $J = 3.2$ and 1.2 Hz, 1H, H-2A), 3.00 (d, $J = 3.6$ Hz, 1H, H-3B), 2.98 (d, $J = 3.6$ Hz, 1H, H-3A), 2.80 (quintet, $J = 7.2$ Hz, 1H, H-7B), 2.34–2.43 (m, 2H, H-6B and H-7A), 2.10–2.16 (m, 1H, H-6A), 1.27 (d, $J = 7.6$ Hz, 3H, CH$_3$-11A), 1.14–1.18 (m, 2H, CH$_2$-5A), 1.12 (s, 6H, CH$_3$-10A, CH$_3$-10B), 1.11 (d, $J = 7.2$ Hz, 3H, CH$_3$-11B), 1.03 (s, 3H, CH$_3$-9A), 0.99 (s, 3H, CH$_3$-9B), 0.81 (dd, $J = 14.0$ and 13.2 Hz, 1H, one of CH$_2$-5B), $^{13}$C NMR (100MHz, CDCl$_3$): 179.08 (C-8B), 179.08 (C-8A), 74.18 (C-1A), 73.56 (C-B), 61.25 (C-3A), 61.17 (C-3B), 52.49 (C-2B), 52.31 (C-2A), 41.47 (C-7B), 39.99 (C-7A), 37.63 (C-6B), 35.79 (C-5B), 34.43 (C-6A), 33.23 (C-5A), 30.05 (C-10A), 28.83 (C-10B), 28.34
7-Hydroxy-4,4,6,7-tetramethyl-9-oxabicyclo[4.3.0]nonan-8-one (5). $^1$H NMR (400 MHz, CDCl$_3$): 5.86 (d, $J = 10.4$ Hz, 1H, H-3), 5.76 (dd, $J = 10.4$ and 5.6 Hz, 1H, H-2), 4.62 (d, $J = 5.2$ Hz, 1H, H-1), 1.24–1.32 (m, 1H, one of CH$_2$-5), 1.28 (s, 3H, CH$_3$-12), 1.14 (s, 3H, CH$_3$-11), 1.12 (s, 3H, CH$_3$-10), 1.11–1.14 (m, 1H, one of CH$_2$-5), 1.01 (s, 3H, CH$_3$-9). $^{13}$C NMR (100 MHz, CDCl$_3$): 177.75 (C-8B), 144.45 (C-3), 117.48 (C-2), 78.46 (C-1), 43.06 (C-7), 40.02 (C-5), 33.28 (C-10), 32.32 (C-6), 28.90 (C-9), 29.74 (C-4), 17.07 (C-12), 14.71 (C-11). (Supplementary Materials, Figures S19–S22).

2,3-Epoxy-4,4,6,7-tetramethyl-9-oxabicyclo[4.3.0]nonan-8-one (6). $^1$H NMR (400 MHz, CDCl$_3$): 4.23 (s, 1H, H-1), 3.27 (d, $J = 3.6$ Hz, 1H, H-2), 2.93 (d, $J = 3.2$ Hz, 1H, H-3), 2.75 (q, $J = 7.2$ Hz, 1H, H-7), 1.32 (d, $J = 15.2$ Hz, 1H, one of CH$_2$-5), 1.23 (d, $J = 15.2$ Hz, 1H, one of CH$_2$-5), 1.16 (s, 3H, CH$_3$-12), 1.10 (s, 3H, CH$_3$-10), 1.05 (d, $J = 7.2$ Hz, 3H, CH$_3$-11), 0.93 (s, 3H, CH$_3$-9). $^{13}$C NMR (100 MHz, CDCl$_3$): 178.28 (C-8), 79.32 (C-1), 60.62 (C-3), 53.81 (C-2), 40.04 (C-4), 39.66 (C-7), 38.18 (C-5), 33.07 (C-6), 29.52 (C-10), 26.59 (C-12), 23.66 (C-11), 7.44 (C-9). (Supplementary Materials, Figures S23–S26).

2,3-Epoxy-7-hydroxy-4,4,6,7-tetramethyl-9-oxabicyclo[4.3.0]nonan-8-one (7). White solid, m.p. = 109–110 °C. $^1$H NMR (400 MHz, CDCl$_3$): 4.77 (s, 1H, H-1), 3.58 (dd, $J = 3.6$ and 2.4 Hz, 1H, H-2), 2.96 (dd, $J = 3.6$ and 1.2 Hz, 1H, H-3), 1.22 (s, 3H, CH$_3$-11), 1.21 (s, 3H, CH$_3$-12), 1.20 (s, 3H, CH$_3$-10), 1.07 (s, 3H, CH$_3$-9), 1.04 (d, $J = 14.0$ Hz, 1H, one of CH$_2$-5), 0.80 (d, $J = 14.0$ Hz, 1H, one of CH$_2$-5). $^{13}$C NMR (100 MHz, CDCl$_3$): 176.73 (C-8B), 144.45 (C-3), 117.48 (C-2), 78.46 (C-1), 43.06 (C-7), 40.02 (C-5), 33.28 (C-10), 32.32 (C-6), 28.90 (C-9), 29.74 (C-4), 17.07 (C-12), 14.71 (C-11). (Supplementary Materials, Figures S33–S38).

4.4. Bioassay

Tests were made using the strains of the bacteria, Escherichia coli C1 and Staphylococcus aureus; yeast, Candida albicans KL-1; and filamentous fungi, Alternaria alternata, Fusarium linii A3, and Aspergillus niger XP, which came from the collection of the Department of Biotechnology and Food Microbiology, Wroclaw University of Environmental and Life Sciences. These tests were performed on the automated Bioscreen C system (Automated Growth Curve Analysis System, Lab Systems, Turku, Finland) following the procedure described below. The bacterial cultures were cultivated in a liquid broth containing 1.5% of dry bullion (Biocorp, Warszawa, Poland) and 1% of glucose dissolved in distilled water for 48 h. Yeast and fungi were cultured in YPG medium containing 1% of yeast extract, 1% of bacteriological peptone, and 1% of glucose dissolved in distilled water for 48 and 96 h, respectively. The mixture of 280 µL of culture medium and 10 µL of cells or spore solution (final density $1 \times 10^6$ cells mL$^{-1}$) was placed in the 300 µL wells of Bioscreen plates. Tested lactones after being dissolved in 10 µL dimethyl sulphoxide were used at a final concentration of 0.1% (v/v). The temperatures were controlled to 25 °C for filamentous fungi and 30 °C for bacteria and yeasts. The cell cultures were placed on a continuous shaker and the optical densities of the cell suspensions were measured automatically at 560 nm every 30 mins for 2 to 4 days. Each culture was performed in three replicates. Data was analyzed using spreadsheet software (Excel 2010) and the means for three replicates of each type of culture medium were calculated. On the basis of mean values, growth curves were generated...
for each test strain. These curves were compared to the curves obtained for control cultures consisting of medium with the addition of dimethyl sulfoxide.

For all cases in which complete growth inhibition of microorganisms in the screening procedure was observed ($\Delta$OD = 0), the minimal inhibitory concentration (MIC) values were specified. MIC was determined by the dilution method. Sterile 100-well microtiter plates with a nutrient broth with glucose (bacteria) or YM (yeast, filamentous fungi) medium (5% of bacteriological peptone, 3% of yeast extract, 3% of malt extract, 10% of glucose and 20% of agar) were inoculated with a standardized cell suspension of fungal spores of $1 \times 10^5$ cell/mL, and supplemented with suitable lactones at the concentration range from 0.0075 to 0.1 mg/mL, prepared with two-fold serial dilution. The analysis was conducted in three replicates. The plates were incubated at 25°C for 72 h. The results were measured in Bioscreen C at a 560 nm wavelength. The minimal inhibitory concentration was defined as the lowest concentration of a tested compound that completely restricted the growth of the microorganism. In order to verify the obtained results, microbial cultures subjected to each concentration of lactones were inoculated onto nutrient broth with glucose or YM plates and incubated as described above.

5. Conclusions

Two known unsaturated lactones (1 and 2) were biotransformed in cultures, seven strains of the *Pleurotus* genus. In case of lactone 1, the formation of a new hydroxylactone 3 and a known epoxy lactone 4 was observed. Lactone 2 was transformed into the known hydroxylactone 5 and epoxy lactone 6, and two new epoxy-hydroxylactones 7 and 8. During the research, it turned out, that fungi of the *Pleurotus* genus were capable of completely converting the substrate within 7 days. As it was mentioned earlier, filamentous fungi of the genus *Fusarium*, *Penicillium*, *Absidia*, and *Syncphalastrum* needed 14 days for this. Lactones 3 and 4 were formed with high enantiomeric excess, and lactones 5, 6, and 8 with small excess; however, all with the predominance of (−)-enantiomer. Both the unsaturated lactones and hydroxy- and epoxy-derivatives were capable of completely inhibiting the growth of two strains of *A. alternata* and *F. limii*.

Supplementary Materials: The following are available online at http://www.mdpi.com/2073-4344/9/8/643/s1. Figure S1. $^1$H NMR (400 MHz, CDCl$_3$) spectrum of unsaturated lactone 1, Figure S2. $^{13}$C NMR (100 MHz, CDCl$_3$) spectrum of unsaturated lactone 1, Figure S3. $^1$H NMR (400 MHz, CDCl$_3$) spectrum of unsaturated lactone 2, Figure S4. $^{13}$C NMR (100 MHz, CDCl$_3$) spectrum of unsaturated lactone 2, Figure S5. $^1$H NMR (400 MHz, CDCl$_3$) spectrum of hydroxylactone 3, Figure S6. COSY (100 MHz, CDCl$_3$) spectrum of hydroxylactone 3, Figure S7. HMQC (100 MHz, CDCl$_3$) spectrum of hydroxylactone 3, Figure S8. $^{13}$C NMR (100 MHz, CDCl$_3$) spectrum of hydroxylactone 3, Figure S9. HRMS spectrum of hydroxylactone 3, Figure S10. IR spectrum of hydroxylactone 3 Figure S11. $^1$H NMR (400 MHz, CDCl$_3$) spectrum of epoxylactone 4-A+B, Figure S12. COSY (100 MHz, CDCl$_3$) spectrum of epoxylactone 4-A+B, Figure S13. HMQC (100 MHz, CDCl$_3$) spectrum of epoxylactone 4-A+B, Figure S14. $^{13}$C NMR (100 MHz, CDCl$_3$) spectrum of epoxylactone 4-A+B, Figure S15. $^1$H NMR (400 MHz, CDCl$_3$) spectrum of epoxylactone 4-B, Figure S16. COSY (100 MHz, CDCl$_3$) spectrum of epoxylactone 4-B, Figure S17. HMQC (100 MHz, CDCl$_3$) spectrum of epoxylactone 4-B, Figure S18. $^{13}$C NMR (100 MHz, CDCl$_3$) spectrum of epoxylactone 4-B, Figure S19. $^1$H NMR (400 MHz, CDCl$_3$) spectrum of hydroxylactone 5, Figure S20. COSY (100 MHz, CDCl$_3$) spectrum of hydroxylactone 5, Figure S21. HMQC (100 MHz, CDCl$_3$) spectrum of hydroxylactone 5, Figure S22. $^{13}$C NMR (100 MHz, CDCl$_3$) spectrum of hydroxylactone 5, Figure S23. $^1$H NMR (400 MHz, CDCl$_3$) spectrum of epoxy lactone 6, Figure S24. COSY (100 MHz, CDCl$_3$) spectrum of epoxy lactone 6, Figure S25. HMQC (100 MHz, CDCl$_3$) spectrum of epoxy lactone 6, Figure S26. $^{13}$C NMR (100 MHz, CDCl$_3$) spectrum of epoxy lactone 6, Figure S27. $^1$H NMR (400 MHz, CDCl$_3$) spectrum of epoxy-hydroxylactone 7, Figure S28. COSY (100 MHz, CDCl$_3$) spectrum of epoxy-hydroxylactone 7, Figure S29. HMQC (100 MHz, CDCl$_3$) spectrum of epoxy-hydroxylactone 7, Figure S30. $^{13}$C NMR (100 MHz, CDCl$_3$) spectrum of epoxy-hydroxylactone 7, Figure S31. HRMS spectrum of epoxy-hydroxylactone 7, Figure S32. IR spectrum of epoxy-hydroxylactone 7, Figure S33. $^1$H NMR (400 MHz, CDCl$_3$) spectrum of epoxy-hydroxylactone 8, Figure S34. COSY (100 MHz, CDCl$_3$) spectrum of epoxy-hydroxylactone 8, Figure S35. HMQC (100 MHz, CDCl$_3$) spectrum of epoxy-hydroxylactone 8, Figure S36. $^{13}$C NMR (100 MHz, CDCl$_3$) spectrum of epoxyhydroxylactone 8, Figure S37. HRMS spectrum of epoxy-hydroxylactone 8, Figure S38. IR spectrum of epoxy-hydroxylactone 8 Figure S39. Chiral chromatogram of hydroxylactone 3, Figure S40. Chiral chromatogram of epoxy lactone 4, Figure S41. Chiral chromatogram of hydroxylactone 5, Figure S42. Chiral chromatogram of epoxy lactone 6, Figure S43. Chiral chromatogram epoxy-hydroxylactone 8, Figure S44. The assessment of the effects of the lactones 1–8 on the growth of *E. coli*, Figure S45. The assessment of the effects of the lactones 1–8 on the growth of *S. aureus*, Figure S46. The assessment
of the effects of the lactones 1–8 on the growth of C. albicans, Figure S47. The assessment of the effects of the lactones 1–8 on the growth of Alternaria sp., Figure S48. The assessment of the effects of the lactones 1–8 on the growth of F. linii, Figure S49. The assessment of the effects of the lactones 1–8 on the growth of A. niger.

Author Contributions: M.C. conceived and designed the experiments and analyzed the data, W.M., K.W. and A.K. performed the experiments; B.Z. performed the biological analysis, G.M. performed the high resolution mass spectrometry analysis, E.G. takes care of the mushroom collection, E.G. and S.P. contributed to the discussion of results. M.G., W.M. and K.W. wrote the paper. All authors read and approved the final manuscript.

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