Synthesis and Biological Activity Screening of Newly Synthesized Trimethoxyphenyl-Based Analogues as Potential Anticancer Agents

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Abstract: A group of novel trimethoxyphenyl (TMP)-based analogues were synthesized by varying the azalactone ring of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene)oxazolone 1 and characterized using NMR spectral data as well as elemental microanalyses. All synthesized compounds were screened for their cytotoxic activity utilizing the hepatocellular carcinoma (HepG2) cell line. Compounds 9, 10, and 11 exhibited good cytotoxic potency with IC₅₀ values ranging from 1.38 to 3.21 µM compared to podophyllotoxin (podo) as a reference compound. In addition, compounds 9, 10, and 11 exhibited potent inhibition of β-tubulin polymerization. DNA flow cytometry analysis of compound 9 shows cell cycle disturbance at the G2/M phase and a significant increase in Annexin-V-positive cells compared with the untreated control. Compound 9 was further studied regarding its apoptotic potential in HepG2 cells; it decreased the level of MMP and Bcl-2 as well as boosted the level of p53 and Bax compared with the control HepG2 cells.

Keywords: diamide; oxazolone; imidazolone; triazinone; cytotoxicity; tubulin; cell cycle analysis; apoptosis

1. Introduction

Target-based chemotherapy is designed to exploit specific molecular targets within the body [1,2]. The ultimate goal of this type of therapy is to develop alternative and more effective drugs that specifically destroy the specific cellular target without having a significant effect on other cellular structures [3,4]. Microtubules are the most important validated target for the treatment of malignancies [5–7]. Microtubules, built from α,β-tubulin building units, are intimately involved in many diverse cellular structures as well
as the replication of the cells and cell division [8,9]. During cell division, the microtubules required for mitosis begin to grow and form structures known as mitotic spindles [10]. These mitotic spindles begin to separate newly formed chromosomes to give new daughter cells [11]. Inhibition or decaying of the microtubule formation in tumor cells leads to the failure of formation of mitotic spindled and separation of chromosomes, which, eventually, inhibits cells’ reproduction and growth [12,13]. Thus, agents that interfere with the dynamics of tubulin may also act as inhibitors of cell division [14].

Many natural products possessing Trimethoxy Phenyl (TMP) rings, e.g., colchicine (col) I and podophyllotoxin (podo) II, were found to be potent cytotoxic agents that exert their cytotoxic properties by interfering with the tubulin-microtubule system [15,16]. Therefore, there has been considerable interest in the development of new anti-tubulin molecules based on these natural compounds as structural leads [17]. Sinensetin III is a plant-derived pentamethoxy flavone compound. It possesses potent cytotoxic properties toward a variety of cancer cell lines involving interaction with breast cancer resistance protein (ABCG2/BCRP) and antiangiogenesis [18] (Figure 1).

![Chemical structures](image)

**Figure 1.** Chemical structure of col I, podo II, Sinensetin III, Imidazole IV, and 1,2,4-triazine V as anti-cancer agents and some reported tubulin polymerization inhibitors, VI and VII.

It was reported that diamide, imidazole, and 1,2,4-triazine derivatives as a template for antiproliferative agents have cytotoxic activities over several cancer cell lines [19–21]. The average IC₅₀ of imidazole RABI-112 IV over eight cancer cell lines was 14 nM, and it caused cells’ arrest at the G2/M phase of the cell cycle besides its detrimental effect on microtubule networks [22]. In addition, 1,2,4-triazine-chalcone hybrid V displayed significant antiproliferative activity against A-594, EC-109, PC-3, HCT-116, and MGC-803.
Cell lines with IC₅₀ values of 0.52, 0.78, 0.61, 0.43, and 0.41 µM, respectively, compared with 5-fluorouracil (5-FU) as a reference control [23]. Additionally, compounds VI and VII contain dimethoxy phenyl (DMP) and TMP rings that exhibit cytotoxic potency over different cancer cell lines and promising tubulin polymerization inhibition activities [24,25] (Figure 1).

Based on the above structure analysis, the current study is concerned with the design and synthesis of new compounds decorated with DMP and TMP moieties starting from 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene)oxazolone 1. The prepared derivatives were subjected to in vitro cytotoxic evaluation utilizing a HepG2 cell line followed by a tubulin inhibition profile. Moreover, tubulin inhibition in HepG2 cells was found to trigger an apoptotic cascade. So that the apoptotic effects of the most potent compound were investigated through cell cycle analysis, inhibition of mitochondrial membrane potential, and determination of p53, Bax, and Bcl-2 levels.

2. Results and Discussion

2.1. Chemistry

The synthetic strategy to prepare the target molecules is depicted in Scheme 1. The starting oxazolone was prepared in a good yield via reaction of 2-(3,4-dimethoxybenzamido)acetic acid with 3,4,5-trimethoxybenzaldehyde in acetic anhydride containing anhydrous sodium acetate according to a previously reported method [26]. Compound 1 was then subjected to a ring-opening reaction in refluxing methanol or absolute ethanol in the presence of triethylamine (Et₃N) as a base catalyst to provide methyl or ethyl acrylate ester 2a,b, respectively. In addition, the interaction of oxazolone with respective amino acids, namely, glycine or alanine, in refluxing pure ethanol gave the corresponding carboxylic acid derivatives 4a and b, respectively. Additionally, the synthesis of diamide molecules 4a–c took place by heating to reflux a mixture of the key starting material 1 with appropriate aromatic amine, namely, m-toluidine, p-aminophenol, or p-anisidine, in glacial acetic acid for 1–2 h. Furthermore, compound 5 was obtained via reaction of the key starting material 1 with phenyl hydrazine at room temperature for 3 h. Moreover, refluxing the oxazolone 1 with isonicotinic acid hydrazide in pure ethanol resulted in the formation of pyridine carbohydrazide derivative 6. On the other hand, cyclocondensation of the oxazolone 1 with respective aryl amine in glacial acetic acid in the presence of anhydrous sodium acetate furnished the corresponding N-aryl imidazolone derivatives 7a and b. The preparation of N-thioamide-1,2,4-triazinone derivative 8 was achieved by the reaction of oxazolone 1 with thiosemicarbazide in glacial acetic acid in the presence of anhydrous sodium acetate, while its reaction with phenylhydrazine under reflux yielded the N-phenyl-1,2,4-triazinone molecule 9. Moreover, the N-pyridoyl-1,2,4-triazinone derivative 10 was prepared by condensing oxazolone 1 with isonicotinic acid hydrazide in refluxing glacial acetic acid in the presence of anhydrous sodium acetate. Finally, N-phenylthiazole-1,2,4-triazinone 11 was achieved via the reaction of the key starting material 1 with 2-hydrazenyl-4-phenylthiazole in dry DMF containing a few drops of glacial acetic acid.

2.2. Biology

2.2.1. Cytotoxic Activity against Liver HepG2 Cell Line

All the newly synthesized analogs were evaluated for their in vitro cytotoxicity against the hepatocellular carcinoma HepG2 cell line. Podo was utilized as a reference control in the current study. The results over HepG2 cells revealed that compounds 9, 10, and 11 exhibited potent cytotoxic activity with IC₅₀ ranges within 1.38–3.21 µM. The N-phenyl triazinone derivative 9 (IC₅₀ = 1.38 µM) has proved to be the most potent followed by the N-pyridoyl triazinone 10 (IC₅₀ = 2.52 µM) and the N-phenylthiazoly triazinone 11 (IC₅₀ = 3.21 µM). In addition, compounds 3a, 5, 6, 7a, 7b, and 8 displayed moderate cytotoxic activity with an IC₅₀ concentration of less than 10 µM. Compounds 2a, 2b, 3b, and 4b showed relatively weak cytotoxic activity with an IC₅₀ concentration of less than 20 µM. The N-phenyl triazinone derivative 9 was further investigated for its cytotoxic effects against normal liver
cell line HL-7702. The results showed a selective mode of cytotoxic effect with an IC₅₀ value of 29.07 µM. These results indicate the selectivity of N-phenyl triazinone 9 for tumor HepG2 cells and its selectivity for the normal liver cell line (Table 1).

Scheme 1. Synthesis of the target compounds 2–11. Reagents and conditions: (i) Et₃N, respective alcohol; (ii) respective amino acid, Et₃N, EtOH; (iii) respective aryl amine, NaOAc, AcOH; (iv) phenyl hydrazine, EtOH; (v) isonicotinic acid hydrazide, EtOH; (vi) respective aryl amine, NaOAc, AcOH; (vii) thiosemicarbazide, NaOAc, AcOH; (viii) phenyl hydrazine, EtOH; (ix) isonicotinic acid hydrazide, NaOAc, AcOH; (x) 2-hydrazinyl-4-phenylthiazole, AcOH, DMF.

2.2.2. Tubulin Polymerization Inhibition Activity

To confirm whether the observed cytotoxic potency of our designed compounds was correlated to β-tubulin inhibition activity, the inhibitory effect of the most cytotoxic compounds 9, 10, and 11 on the tubulin polymerization was carried out using podo as a positive control. The selected candidates were tested at a concentration equal to their IC₅₀ concentration dose value. ELISA analysis results revealed that compounds 9 and 10 displayed potent β-tubulin polymerization inhibition with percent inhibition of 86.73% and 80.51%, respectively, compared to podo (91.32% β-tubulin polymerization inhibition). On the other hand, β-tubulin polymerization inhibition percentage for compound 11 was recorded as 69.95%. It could be assumed from these results that the molecular target of these analogs is indeed tubulin (Figure 2).
Table 1. Cytotoxic screening of the tested trimethoxyphenyl-based analogues 2a–11. Values are the mean ± SE.

| Comp No | IC_{50} Value (µM)   |
|---------|----------------------|
|         | HepG2                | HL-7702              |
| 2a      | 14.39 ± 0.82         | NT                   |
| 2b      | 11.09 ± 0.77         | NT                   |
| 3a      | 8.13 ± 0.54          | NT                   |
| 3b      | 19.41 ± 1.43         | NT                   |
| 4a      | 45.53 ± 2.89         | NT                   |
| 4b      | 17.69 ± 1.10         | NT                   |
| 4c      | 23.72 ± 1.37         | NT                   |
| 5       | 9.24 ± 0.52          | NT                   |
| 6       | 8.74 ± 0.57          | NT                   |
| 7a      | 4.53 ± 0.29          | NT                   |
| 7b      | 6.45 ± 0.24          | NT                   |
| 8       | 4.02 ± 0.09          | NT                   |
| 9       | 1.38 ± 0.15          | 29.07 ± 2.03         |
| 10      | 2.52 ± 0.24          | NT                   |
| 11      | 3.21 ± 0.22          | NT                   |
| podo    | 2.08 ± 0.04          | 18.59 ± 0.58         |

NT: not tested.

Figure 2. Compounds 9, 10, and 11 inhibited β-tubulin in HepG2 cells. (A) HepG2 cells were treated with 1.38, 2.52, 3.21, and 2.08 µM of compounds 9, 10, 11, and podo, respectively, and tubulin percentage was determined by ELISA assay. (B) β-Tubulin polymerization inhibition percentage induced by compounds 9, 10, and 11 compared to podo at their IC_{50} concentration (µM).

2.2.3. Cell Cycle Analysis of Compound 9

It is well documented that exposure to microtubule targeting agents leads to the malformed mitotic spindle which eventually leads to cell cycle arrest at the G2/M phase and apoptosis induction [27]. Thus, the effect of the selected candidate 9 on the cell cycle progression of the HepG2 cell line was examined by flow cytometry analysis (FACS). The HepG2 cell line was treated with the IC_{50} concentration of compound 9 for 48 h. The FACS analysis results reveal that the selected compound 9 resulted in G2/M phase arrest as denoted by the increase in the G2/M cell population percentage (30.83%) compared with the control sample (13.28%) with about a 3-fold increase compared to the untreated control sample. In conclusion, compound 9 showed HepG2 inhibition of cell growth via cell cycle arrest at the G2/M phase (Figure 3).
2.2.4. Apoptosis-Inducing Activity of Compound 9

The extent of apoptosis induced by compound 9 in HepG2 cells was quantified by the FACS analysis. Cells were harvested after 48 h and analyzed for apoptosis percentage by flow cytometry. Figure 4 showed the percentage of cells in each stage of apoptosis induced by compound 9 at a concentration equal to its IC$_{50}$ dose value compared with the untreated sample. The results obtained for compound 9 showed a large increase in cells in the early and late stages after 48 h of exposure. It could be noticed that compound 9 increases the percentage of early and late apoptosis by 16.63- and 60.11-fold, respectively, compared with the untreated control. These results indicate that the compound’s mechanism of action may occur via the apoptosis pathway mechanism.

2.2.5. Mitochondrial Membrane Potential

Mitochondrial Membrane Potential (MMP) plays a vital role in the initiation of the apoptosis process [28]. Loss of MMP is characteristic of an early stage of apoptosis [29]. In the current study, HepG2 cells were treated with compound 9 at a concentration level equal to its IC$_{50}$ concentration (1.38 μM), and MMP was determined using FACS analysis. The
results indicate that compound 9 significantly reduced MMP compared with the untreated control HepG2 cells. From Figure 5, it is noticeable that compound 9 decreases the level of MMP by 3.12-fold compared with negative HepG2 cells. These results reveal that compound 9 exerted its cytotoxic activity by promoting MMP dissipation in the G2/M phase in HepG2 cells.

**Figure 5.** Compound 9 promotes MMP dissipation in HepG2 cells. (A) HepG2 cells were treated with 1.38 µM of compound 9 for 48 h. MMP was quantized using image-based flow cytometry. (B) The quantification of MMP in HepG2 cells after treatment with compound 9 at its IC\(_{50}\) concentration (µM).

2.2.6. In Vitro ELISA Quantification of the Level of p53, Bax and Bcl-2

Many studies suggested that tubulin-inhibiting agents cause G2/M phase arrest followed by apoptosis induction [30]. Compound 9 was screened for its ability to activate p53 and Bax as well as downregulate the level of Bcl-2 to investigate the mechanistic apoptotic pathway that catalyzed the apoptosis process [30]. HepG2 cells were treated with compound 9 at a concentration equal to its IC\(_{50}\) in µM, and the level of p53, Bax, and Bcl-2 was quantized using ELISA analysis. The results obtained for compound 9 show the increase in the level of p53 and Bax by 8.91- and 4.50-fold, respectively, compared with the negative control samples. Similarly, compound 9 showed a decrease in the level of Bcl-2 by 5.58-fold less than the negative control samples. The results indicate that compound 9’s mechanisms of action induce the intrinsic pathway of the apoptotic mechanism through increasing the level of p53 and Bax, together with decreasing the level of Bcl-2 (Figure 6).

2.2.7. Molecular Docking Study

N-phenyl triazinone 9 was docked into the validated binding site of the tubulin crystal structure (PDB code: 5LYJ). The N-phenyl triazinone ligand interacted with the active site of 5LYJ by four hydrogen bonds. N-phenyl triazinone molecule 9 interacted with its NH group of triazinone rings as a hydrogen bond donor with amino acid Asn 101 residue. Additionally, N-phenyl triazinone molecule 9 interacted with its TMP ring as a hydrogen bond acceptor by three hydrogen bonds with amino acids Gln 11 and Asn 206. Furthermore, the docking score achieved by N-phenyl triazinone molecule 9 was −14.67 kcal/mol (Figure 7). In conclusion, the interaction results are consistent with the obtained results of tubulin polymerization inhibitory activity and justify the observed experimental findings.
Figure 6. (A) The quantification of p53. (B) The quantification of Bax. (C) The quantification of Bcl-2. HepG2 cells were treated with 1.38 µM of compound 9 for 48 h and apoptotic markers were determined by ELISA assay.

Figure 7. The binding poses (2D and 3D) of N-phenyl triazinone 9 in tubulin active site (PDB ID: 5LYJ).

3. Experimental
3.1. Chemistry: General

Melting points were determined in open capillary tube using Electrothermal Digital melting point apparatus and were uncorrected. $^1$H-NMR and $^{13}$C-NMR spectra were obtained with a Bruker 400 MHz DRX-Avance NMR spectrometer, and peak positions are given in ppm downfield from tetramethylsilane (TMS) as the internal standard. Elemental
analyses were performed on Elementar, Vario El, Microanalytical unit, Cairo, Egypt, and were found within ±0.4% of the theoretical values. Chemicals and reagents were obtained from commercial sources and used without further purification.

3.2. Chemistry

3.2.1. General Procedure for the Preparation of (Z)-2-(3,4-Dimethoxyphenyl)-4-(3,4,5-Trimethoxybenzamido)oxazol-5(4H)-one (1)

A mixture of 3,4,5-trimethoxybenzaldehyde (2.39 g, 10 mmol), 2-(3,4-dimethoxybenzamido)acetic acid (1.96 g, 10 mmol) and anhydrous sodium acetate (0.90 g, 11 mmol) in acetic anhydride (10 mL) was heated in oil bath at 80 °C for 2 h. After heating time, 100 mL of absolute ethanol was added to the reaction mixture, and the mixture was allowed to stand overnight. The precipitate formed was filtered off, washed with two 25 mL portions of ice-cold ethanol, dried and crystallized from ethanol/water (3:1) to get pure compound 1.

Yellow powder (2.48 g, 54.33%), m.p. 195–197 °C. 1H-NMR (400 MHz, DMSO-d₆, ppm): 3.76 (s, 3H, OCH₃), 3.85 (s, 3H, OCH₃), 3.87 (s, 6H, 2OCH₃), 3.88 (s, 3H, OCH₃), 7.21–7.16 (m, 2H, arom.CH and olefinic CH), 7.56 (d, J = 1.6 Hz, 1H, arom.CH), 7.67 (dd, J = 8.4, 1.7 Hz, 1H, arom.CH), 7.74 (s, 2H, arom.CH). 13C-NMR (100 MHz, DMSO-d₆, ppm): 55.34 (OCH₃), 55.76 (2OCH₃), 55.89 (OCH₃), 60.25 (OCH₃), 109.67 (C2,6 trimethoxybenzylidene), 111.95 (C5 dimethoxyphenyl), 117.00 (C2 dimethoxybenzamide), 122.18 (C6 dimethoxyphenyl), 128.97 (C olefinic), 129.43 (C1 dimethoxybenzamide), 131.25 (C4 oxazolone), 132.17 (C1 trimethoxybenzylidene), 140.14 (C4 trimethoxybenzylidene), 149.00 (C3 dimethoxyphenyl), 152.80 (C4 oxazolone), 153.51 (C3,5 trimethoxybenzylidene), 162.37 (C2 oxazolone), 167.04 (C=O oxazolone). Anal. Calcd. for C₂₁H₂₁NO₇ (399.39): C, 61.15; H, 5.30; N, 3.51. Found: C, 63.08; H, 5.36; N, 3.54.

3.2.2. General Procedure for the Preparation of Ester Derivatives 2a,b

A suspension of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene)oxazolone 1 (399 mg, 1 mmol) in respective alkyl alcohol (25 mL), namely, methyl alcohol or ethyl alcohol, was added to Et₃N (0.5 mL). The resulting mixture was heated to reflux for 4–5 h until complete on thin-layer chromatography. Evaporation of the solvent afforded the crude ester product that was crystallized from appropriate solvent to get pure compound 2a,b.

(Z)-Methyl 2-(3,4-Dimethoxybenzamido)-3-(3,4,5-Trimethoxyphenyl)acrylate (2a)

White powder (273 mg, 63.41%), crystallized from methanol/H₂O (3:1), m.p. 248–250 °C. 1H-NMR (400 MHz, DMSO-d₆, δ ppm): 3.66 (s, 6H, 2OCH₃), 3.66 (s, 3H, OCH₃), 3.73 (s, 3H, OCH₃), 3.80 (s, 3H, OCH₃), 3.83 (s, 3H, OCH₃), 7.05–7.12 (m, 3H, arom.CH and olefinic CH), 7.39 (s, 1H, arom.CH), 7.60 (s, 1H, arom.CH), 7.67 (dd, J = 8.4, 1.7 Hz, 1H, arom.CH), 9.93 (s, 1H, NH). 13C-NMR (100 MHz, DMSO-d₆, ppm): 52.22 (OCH₃), 55.62 (OCH₃), 55.67 (OCH₃), 55.71 (OCH₃), 60.11 (2OCH₃), 107.67 (C2,6 trimethoxyphenyl), 110.72 (C5 dimethoxybenzamide), 111.00 (C2 dimethoxybenzamide), 121.18 (C6 dimethoxybenzamide), 125.29 (C olefinic), 125.84 (C1 trimethoxybenzamide), 128.79 (C1 dimethoxybenzamide), 133.27 (C olefinic), 138.69 (C4 trimethoxyphenyl), 148.35 (C3 dimethoxybenzamide), 151.86 (C5 dimethoxybenzamide), 162.52 (C3,5 trimethoxybenzylidene), 165.36 (C=O aci-dide), 165.68 (C=O dimethoxybenzamide). Anal. Calcd. for C₂₂H₂₅NO₇ (431.44): C, 61.25; H, 5.84; N, 3.25. Found: C, 61.29; H, 5.91; N, 3.18.

(Z)-Ethyl 2-(3,4-Dimethoxybenzamido)-3-(3,4,5-Trimethoxyphenyl)acrylate (2b)

White powder (267 mg, 59.92%), crystallized from ethanol/H₂O (3:1), m.p. 248–250 °C. 1H-NMR (400 MHz, DMSO-d₆, δ ppm): 1.25 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 3.67 (s, 6H, 2OCH₃), 3.67 (s, 3H, OCH₃), 3.81 (s, 3H, OCH₃), 3.84 (s, 3H, OCH₃), 4.20 (q, J = 7.1 Hz, 2H, OCH₂CH₃), 7.08 (s, 2H, arom.CH), 7.10 (s, 1H, olefinic CH), 7.38 (s, 1H, arom.CH), 7.56 (s, 1H, arom.CH), 7.64–7.70 (m, 1H, arom.CH), 9.92 (s, 1H, NH). 13C-NMR (100 MHz, DMSO-d₆, δ ppm): 14.19 (OCH₂CH₃), 55.60 (OCH₃), 55.65 (OCH₃), 55.70 (2OCH₃), 60.09 (OCH₃), 60.80 (OCH₂CH₃), 107.62 (C2,6 trimethoxyphenyl), 110.71 (C5 dimethoxybenzamide), 111.00
Molecules 2022, 27, 4621

(C2 dimethoxybenzamide), 121.13 (C6 dimethoxybenzamide), 125.42 (C olefinic), 126.16 (C1 trimethoxyphenyl), 128.84 (C1 dimethoxybenzamide), 132.96 (C olefinic), 138.61 (C4 dimethoxyphenyl), 148.33 (C3 dimethoxybenzamide), 151.82 (C4 dimethoxybenzamide), 152.60 (C3,5 trimethoxyphenyl), 163.96 (C=O amide), 165.50 (C=O dimethoxybenzamide).

111.17 (C2 dimethoxybenzamide), 121.13 (C6 dimethoxybenzamide), 125.42 (C olefinic), 126.16 (C1 trimethoxyphenyl), 128.84 (C1 dimethoxybenzamide), 132.96 (C olefinic), 138.61 (C4 dimethoxyphenyl), 148.33 (C3 dimethoxybenzamide), 151.82 (C4 dimethoxybenzamide), 152.60 (C3,5 trimethoxyphenyl), 163.96 (C=O amide), 165.50 (C=O dimethoxybenzamide).

129.87 (C olefinic), 137.94 (C4 trimethoxyphenyl), 148.28 (C3 dimethoxybenzamide), 151.82 (C4 dimethoxybenzamide), 152.56 (C3,5 trimethoxyphenyl), 163.96 (C=O amide), 165.50 (C=O dimethoxybenzamide).

Acid (Acids)

3.2.3. General Procedure for the Preparation of 2-(2-(3,4-Dimethoxybenzamido)-3-(3,4,5-Trimethoxyphenyl)acrylamido)carboxylic Acids 3a,b

To a suspension of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene)oxazolone 1 (399 mg, 1 mmol) in absolute ethanol, we added respective amino acid (1 mmol), Et3N (10 drops), and the reaction mixture was heated to reflux for 10 h. After completion of the reaction, water was added (100 mL) and the organic layer was extracted with EtOAc. The organic layer was evaporated under reduced pressure. The crude product was purified by crystallization from ethanol/H2O (3:1) to afford pure compound 3a,b.

(Z)-2-(2-(3,4-Dimethoxybenzamido)-3-(3,4,5-Trimethoxyphenyl)acrylamido)acetic Acid (3a)

White powder (238 mg, 50.17%), m.p. 248–250 °C. 1H-NMR (400 MHz, DMSO-d6, δ ppm): 3.44 (s, 2H, CH2), 3.59 (s, 6H, 2OCH3), 3.63 (s, 3H, OCH3), 3.80 (s, 3H, OCH3), 3.83 (s, 3H, OCH3), 6.94 (s, 2H, arom.CH), 7.08 (d, J = 8.1 Hz, 1H, arom.CH), 7.34 (s, 1H, olefinic CH), 7.49 (s, 1H, NH), 7.63 (s, 1H, arom.CH), 7.69 (d, J = 8.4 Hz, 1H, arom.CH), 9.95 (s, 1H, NH). 13C-NMR (100 MHz, DMSO-d6, δ ppm): 44.19 (CH2), 55.59 (2OCH3), 55.68 (OCH3), 60.05 (OCH3), 107.03 (C2,6 trimethoxybenzylidene), 110.92 (C5 dimethoxybenzamide), 111.04 (C2 dimethoxybenzamide), 121.36 (C6 dimethoxybenzamide), 125.58 (C olefinic), 129.21 (C1 trimethoxyphenyl), 129.59 (C1 dimethoxybenzamide), 129.87 (C olefinic), 137.94 (C4 trimethoxybenzylidene), 148.28 (C3 dimethoxybenzamide), 151.82 (C4 dimethoxybenzamide), 152.56 (C3,5 trimethoxyphenyl), 163.96 (C=O amide), 165.50 (C=O dimethoxybenzamide).

(Z)-2-(2-(3,4-Dimethoxybenzamido)-3-(3,4,5-Trimethoxyphenyl)acrylamido)propanoic Acid (3b)

White powder (236 mg, 48.22%), m.p. 248–250 °C. 1H-NMR (400 MHz, DMSO-d6, δ ppm): 1.33 (d, J = 7.2 Hz, 3H, CH3), 3.60 (s, 6H, 2OCH3), 3.64 (s, 3H, OCH3), 3.80 (s, 3H, OCH3), 3.83 (s, 3H, OCH3), 4.34 (p, J = 7.3 Hz, 1H, CH), 6.94 (s, 2H, arom.CH), 7.07 (d, J = 8.4 Hz, 1H, arom.CH), 7.25 (s, 1H), 7.63 (s, 1H, arom.CH), 7.69 (d, J = 8.2 Hz, 1H, arom.CH), 8.21 (d, J = 7.3 Hz, 1H, NH), 9.73 (s, 1H, NH), 12.38 (s, 1H, OH). 13C-NMR (100 MHz, DMSO-d6, δ ppm): 17.28 (CH3), 48.19 (CH), 55.68 (2OCH3), 55.72 (OCH3), 55.76 (OCH3), 60.15 (OCH3), 107.09 (C2,6 trimethoxybenzylidene), 110.93 (C5 dimethoxybenzamide), 111.17 (C2 dimethoxybenzamide), 121.48 (C6 dimethoxybenzamide), 125.87 (C olefinic), 129.08 (C1 trimethoxyphenyl), 129.66 (C1 dimethoxybenzamide), 129.91 (C olefinic), 138.03 (C4 trimethoxybenzylidene), 148.29 (C3 dimethoxybenzamide), 151.83 (C4 dimethoxybenzamide), 152.64 (C3,5 trimethoxyphenyl), 164.97 (C=O amide), 165.45 (C=O dimethoxybenzamide), 174.15 (C=O carboxylic). Anal. Calcd. for C21H28N2O9 (488.49): C, 59.01; H, 5.78; N, 5.73. Found: C, 59.09; H, 5.81; N, 5.66.

3.2.4. General Procedure for the Preparation of (Z)-N-(3-Aryl-3-oxo-1-(3,4,5-Trimethoxyphenyl)prop-1-en-2-yl)-3,4-Dimethoxybenzamides 4a–c

A mixture of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene)oxazolone 1 (399 mg, 1 mmol) and the appropriate aryl amine (1 mmol) was refluxed in glacial acetic acid (20 mL) in the presence of anhydrous sodium acetate (82 mg, 1 mmol) for 1–2 h. After refluxing time, the reaction mixture was cooled down, poured into crushed ice and filtered. The obtained solid was crystallized from DMF/H2O (1:1) to afford pure compound 4a–c.
(Z)-3,4-Dimethoxy-N-(3-oxo-3-(M-tolylamino)-1-(3,4,5-Trimethoxyphenyl)prop-1-en-2-yl)benzamide (4a)

White powder (279 mg, 55.09%), m.p. 248–250 °C. 1H-NMR (400 MHz, DMSO-d6, δ ppm): 2.28 (s, 3H, CH3), 3.65 (s, 6H, 2OCH3), 3.66 (s, 3H, OCH3), 3.81 (s, 3H, OCH3), 3.83 (s, 3H, OCH3), 6.88 (d, J = 7.5 Hz, 1H, arom.CH), 7.02 (s, 2H, arom.CH), 7.08 (d, J = 8.5 Hz, 1H, arom.CH), 7.15 (s, 1H, olefinic CH), 7.19 (t, J = 7.7 Hz, 1H, arom.CH), 7.54 (d, J = 8.3 Hz, 2H, arom.CH), 7.69 (s, 1H, arom.CH), 7.72 (d, J = 8.6 Hz, 1H, arom.CH), 10.08 (s, 1H, NH), 10.20 (s, 1H, NH). 13C-NMR (100 MHz, DMSO-d6, δ ppm): 21.32 (CH3), 55.78 (OCH3), 55.81 (3OCH3), 60.26 (OCH3), 107.33 (C2,6 trimethoxyphenyl), 111.02 (C5 dimethoxybenzamide), 111.24 (C2 dimethoxybenzamide), 117.57 (C6 tolyl), 120.90 (C6 dimethoxybenzamide), 121.64 (C4 tolyl), 124.32 (C2 tolyl), 125.73 (C olefinic), 128.50 (C5 tolyl), 128.89 (C1 trimethoxyphenyl), 129.75 (C1 dimethoxybenzamide), 130.44 (C olefinic), 137.81 (C3 tolyl), 138.14 (C4 trimethoxyphenyl), 139.23 (C1 tolyl), 148.37 (C3 dimethoxybenzamide), 151.94 (C4 dimethoxybenzamide), 152.73 (C3,5 trimethoxybenzamide), 164.53 (C=O amide), 165.70 (C=O dimethoxybenzamide). Anal. Calcd. for C28H30N2O7 (506.55): C, 66.39; H, 5.97; N, 5.53. Found: C, 66.28; H, 6.05; N, 5.58.

(Z)-N-(3-(4-Hydroxyphenylamino)-3-oxo-1-(3,4,5-Trimethoxyphenyl)prop-1-en-2-yl)-3,4-Dimethoxybenzamide (4b)

White powder (253 mg, 49.77%), m.p. 248–250 °C. 1H-NMR (400 MHz, DMSO-d6, δ ppm): 3.64 (s, 6H, 2OCH3), 3.66 (s, 3H, OCH3), 3.80 (s, 3H, OCH3), 3.83 (s, 3H, OCH3), 6.71 (d, J = 8.8 Hz, 2H, arom.CH), 7.00 (s, 2H, arom.CH), 7.07 (d, J = 8.4 Hz, 1H, arom.CH), 7.16 (s, 1H, olefinic CH), 7.47 (d, J = 8.8 Hz, 2H, arom.CH), 7.65–7.75 (m, 2H, arom.CH), 9.93 (s, 1H, NH), 10.20 (s, 1H, NH). 13C-NMR (100 MHz, DMSO-d6, δ ppm): 55.63 (OCH3), 55.66 (2OCH3), 55.67 (OCH3), 60.09 (OCH3), 107.14 (C2,6 trimethoxyphenyl), 110.84 (C5 dimethoxybenzamide), 111.20 (C2 dimethoxybenzamide), 114.89 (C3,5 aminophenol), 121.49 (C6 dimethoxybenzamide), 122.01 (C2,6 aminophenol), 125.83 (C olefinic), 128.59 (C1 trimethoxyphenyl), 129.78 (C1 dimethoxybenzamide), 130.57 (C olefinic), 130.79 (C1 aminophenol), 137.87 (C4 trimethoxyphenyl), 148.19 (C3 dimethoxybenzamide), 151.71 (C4 dimethoxybenzamide), 152.57 (C3,5 trimethoxybenzamide), 153.63 (C4 aminophenol), 163.91 (C=O amide), 165.43 (C=O dimethoxybenzamide). Anal. Calcd. for C27H28N2O8 (508.52): C, 63.77; H, 5.55; N, 5.51. Found: C, 63.88; H, 5.45; N, 5.38.

(Z)-3,4-Dimethoxy-N-(3-(4-Methoxyphenylamino)-3-oxo-1-(3,4,5-Trimethoxyphenyl)prop-1-en-2-yl)benzamide (4c)

White powder (2.48 g, 58.61%), m.p. 248–250 °C. 1H-NMR (400 MHz, DMSO-d6, δ ppm): 3.64 (s, 6H, 2OCH3), 3.66 (s, 3H, OCH3), 3.73 (s, 3H, OCH3), 3.80 (s, 3H, OCH3), 3.83 (s, 3H, OCH3), 6.89 (d, J = 9.1 Hz, 2H, arom.CH), 7.01 (s, 2H, arom.CH), 7.07 (d, J = 8.5 Hz, 1H, arom.CH), 7.17 (s, 1H, olefinic CH), 7.63 (d, J = 9.0 Hz, 2H, arom.CH), 7.67–7.75 (m, 2H, arom.CH), 10.05 (s, 1H, NH), 10.19 (s, 1H, NH). 13C-NMR (100 MHz, DMSO-d6, δ ppm): 55.16 (OCH3), 55.62 (OCH3), 55.65 (SOCH3), 60.08 (OCH3), 107.15 (C2,6 trimethoxyphenyl), 110.84 (C5 dimethoxybenzamide), 111.17 (C2 dimethoxybenzamide), 113.60 (C3,5 methoxyphenyl), 121.48 (C6 dimethoxybenzamide), 121.75 (C2,6 methoxyphenyl), 125.76 (C olefinic), 128.63 (C1 trimethoxyphenyl), 129.72 (C1 dimethoxybenzamide), 130.48 (C olefinic), 132.42 (C1 methoxyphenyl), 137.90 (C4 trimethoxyphenyl), 148.19 (C3 dimethoxybenzamide), 151.72 (C4 dimethoxybenzamide), 152.56 (C3,5 trimethoxyphenyl), 155.31 (C4 methoxyphenyl), 164.09 (C=O amide), 165.44 (C=O dimethoxybenzamide). Anal. Calcd. for C28H30N2O8 (522.55): C, 64.36; H, 5.79; N, 5.36. Found: C, 64.44; H, 5.72; N, 5.27.
3.2.5. General Procedure for the Preparation of (Z)-3,4-Dimethoxy-N-(3-oxo-3-(2-Phenylhydrazinyl)-1-(3,4,5-Trimethoxyphenyl)prop-1-en-2-yl)benzamide (5)

To a stirred suspension of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene) oxazolone 1 (399 mg, 1 mmol) in absolute ethanol, we added phenyl hydrazine (1 mmol), and the mixture was stirred at room temperature for 3 h. After evaporation under reduced pressure, the obtained crude product was crystallized from ethanol/H₂O (3:1) to get pure compound 5.

White powder (338 mg, 64.73%), m.p. 248–250 °C. ¹H-NMR (400 MHz, DMSO-d₆, δ ppm): 3.61 (s, 6H, 2OCH₃), 3.65 (s, 3H, OCH₃), 3.82 (s, 3H, OCH₃), 6.69 (t, J = 7.3 Hz, 1H, arom.CH), 6.84 (d, J = 7.8 Hz, 2H, arom.CH), 7.00 (s, 2H, arom.CH), 7.07 (d, J = 8.5 Hz, 1H, arom.CH), 7.14 (t, J = 7.8 Hz, 2H, arom.CH), 7.25 (s, 1H, olefinic CH), 7.67 (s, 1H, arom.CH), 7.71 (d, J = 8.5 Hz, 1H, arom.CH), 7.75 (d, J = 2.5 Hz, 1H, NH), 9.85 (s, 1H, NH), 10.03 (d, J = 2.5 Hz, 1H, NH). ¹³C-NMR (100 MHz, DMSO-d₆, δ ppm): 56.08 (2OCH₃), 56.13 (OCH₃), 60.52 (OCH₃), 107.67 (C=O, trimethoxybenzamide), 111.23 (C=O dimethoxybenzamide), 111.74 (C=O dimethoxybenzamide), 112.92 (C=O phenyl), 118.79 (C=O phenyl), 121.97 (C=O dimethoxybenzamide), 126.37 (C=O olefinic), 129.00 (C=O phenyl), 129.18 (C=O dimethoxybenzamide), 129.97 (C=O olefinic), 138.45 (C=O trimethoxybenzamide), 148.57 (C=O dimethoxybenzamide), 149.98 (C=O phenyl), 152.14 (C=O dimethoxybenzamide), 153.02 (C=O trimethoxybenzamide), 156.69 (C=O hydrazide), 166.11 (C=O dimethoxybenzamide). Anal. Calcd. for C₂₇H₂₉N₃O₇ (507.54): C, 63.89; H, 5.76; N, 8.28. Found: C, 64.03; H, 5.87; N, 8.16.

3.2.6. General Procedure for the Preparation of (Z)-N-(3-(2-Isonicotinoylhydrazinyl)-3-oxo-1-(3,4,5-Trimethoxyphenyl)prop-1-en-2-yl)-3,4-Dimethoxybenzamide (6)

To a stirred suspension of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene) oxazolone 1 (399 mg, 1 mmol) in absolute ethanol, we added isonicotinic acid hydrazide (137 mg, 1 mmol). The reaction mixture was refluxed for 5 h. The excess solvent was evaporated under vacuum to afford the crude product. This was further purified by crystallization from ethanol/H₂O (3:1) to furnish pure compound 6.

White powder (285 mg, 53.11%), m.p. 248–250 °C. ¹H-NMR (400 MHz, DMSO-d₆, δ ppm): 3.63 (s, 6H, 2OCH₃), 3.66 (s, 3H, OCH₃), 3.81 (s, 3H, OCH₃), 3.83 (s, 3H, OCH₃), 3.88 (s, 3H, OCH₃), 7.00 (s, 2H, arom.CH), 7.07 (d, J = 8.5 Hz, 1H, arom.CH), 7.18–7.24 (m, 1H, arom.CH), 7.36 (s, 1H, olefinic CH), 7.71 (d, J = 6.0 Hz, 1H, arom.CH), 7.82 (d, J = 6.0 Hz, 2H, arom.CH), 8.77 (d, J = 5.7 Hz, 2H, arom.CH), 9.89 (s, 1H, NH), 10.31 (s, 1H, NH), 10.79 (s, 1H, NH). ¹³C-NMR (100 MHz, DMSO-d₆, δ ppm): 55.63 (2OCH₃), 55.66 (OCH₃), 55.78 (OCH₃), 60.07 (OCH₃), 107.23 (C=O trimethoxybenzamide), 111.25 (C=O dimethoxybenzamide), 112.38 (C=O pyridyl), 121.52 (C=O dimethoxybenzamide), 123.81 (C=O olefinic), 128.10 (C=O hydrazide), 129.28 (C=O dimethoxybenzamide), 130.74 (C=O olefinic), 138.17 (C=O trimethoxybenzylidene), 139.68 (C=O pyridyl), 148.11 (C=O dimethoxybenzamide), 150.40 (C=O pyridyl), 151.70 (C=O dimethoxybenzamide), 152.60 (C=O trimethoxybenzamide), 163.99 (C=O isonicotinic acid hydrazide), 164.61 (C=O amide), 165.56 (C=O dimethoxybenzamide). Anal. Calcd. for C₂₇H₂₉N₃O₇ (536.53): C, 60.44; H, 5.26; N, 10.44. Found: C, 60.61; H, 5.36; N, 10.23.

3.2.7. General Procedure for the Preparation of (Z)-1-Aryl-2-(3,4-Dimethoxyphenyl)-4-(3,4,5-Trimethoxybenzylidene)-1H-Imidazol-5(4H)-ones 7a,b

A suspension of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene) oxazolone 1 (399 mg, 1 mmol) in glacial acetic acid (25 mL) containing anhydrous sodium acetate (98.4 mg, 1.2 mmol) was heated to reflux for 12 h. After refluxing time, the reaction mixture was cooled and poured into crushed ice. The obtained solid product was filtered off and crystallized from ethanol/H₂O (3:1) to afford pure compound 7a,b.
(Z)-2-(3,4-Dimethoxyphenyl)-1-(4-Hydroxyphenyl)-4-(3,4,5-Trimethoxybenzylidene)-1H-Imidazol-5(4H)-one (7a)

White powder (300 mg, 61.08%), m.p. 248–250 °C. \( ^{1} \)H-NMR (400 MHz, DMSO-\( d_{6} \), \( \delta \) ppm): 3.60 (s, 3H, OCH\( _{3} \)), 3.75 (s, 3H, OCH\( _{3} \)), 3.77 (s, 3H, OCH\( _{3} \)), 3.87 (s, 6H, 2OCH\( _{3} \)), 6.87 (d, \( J = 8.7 \) Hz, 2H, arom.CH), 6.96 (d, \( J = 8.7 \) Hz, 1H, arom.CH), 7.03 (dd, \( J = 8.5, 2.0 \) Hz, 1H, arom.CH), 7.11 (s, 1H, olefinic CH), 7.12 (d, \( J = 3.6 \) Hz, 2H, arom.CH), 7.31 (d, \( J = 1.9 \) Hz, 1H, arom.CH), 7.85 (s, 2H, arom.CH), 10.00 (s, 1H, OH). \( ^{13} \)C-NMR (100 MHz, DMSO-\( d_{6} \), \( \delta \) ppm): 54.92 (OCH\( _{3} \)), 55.68 (OCH\( _{3} \)), 55.77 (2OCH\( _{3} \)), 60.24 (OCH\( _{3} \)), 109.69 (C2,6 trimethoxybenzylidene), 111.20 (C5 dimethoxyphenyl), 115.77 (C4 hydroxyphenyl), 116.05 (C3,5 hydroxyphenyl), 120.54 (C1 dimethoxyphenyl), 122.54 (C6 dimethoxyphenyl), 126.05 (C olefinic), 126.10 (C1 trimethoxybenzylidene), 129.42 (C2,6 hydroxyphenyl), 129.93 (C1 hydroxyphenyl), 137.70 (C4 imidazole), 139.49 (C4 trimethoxybenzylidene), 148.15 (C3 dimethoxyphenyl), 151.70 (C4 dimethoxyphenyl), 152.79 (C3,5 trimethoxybenzylidene), 157.77 (C4 hydroxyphenyl), 159.57 (C2 imidazole), 170.41 (C=O imidazole). Anal. Calcd. for C\( _{27} \)H\( _{28} \)N\( _{2} \)O\( _{7} \) (490.50): C, 66.11; H, 5.34; N, 5.71. Found: C, 65.96; H, 5.42; N, 5.84.

(2-(3,4-Dimethoxyphenyl)-1-(4-Methoxyphenyl)-4-(3,4,5-Trimethoxybenzylidene)-1H-Imidazol-5(4H)-one (7b)

White powder (309 mg, 61.26%), m.p. 248–250 °C. \( ^{1} \)H-NMR (400 MHz, DMSO-\( d_{6} \), \( \delta \) ppm): 3.59 (s, 3H, OCH\( _{3} \)), 3.75 (s, 3H, OCH\( _{3} \)), 3.77 (s, 3H, OCH\( _{3} \)), 3.87 (s, 6H, 2OCH\( _{3} \)), 6.96 (s, 1H, arom.CH), 7.01 (dd, \( J = 8.5, 1.8 \) Hz, 1H, arom.CH), 7.07 (d, \( J = 8.8 \) Hz, 2H, arom.CH), 7.13 (s, 1H, olefinic CH), 7.26 (d, \( J = 8.7 \) Hz, 2H, arom.CH), 7.30 (d, \( J = 1.7 \) Hz, 1H, arom.CH), 7.85 (s, 2H, arom.CH). \( ^{13} \)C-NMR (100 MHz, DMSO-\( d_{6} \), \( \delta \) ppm): 54.91 (OCH\( _{3} \)), 55.48 (OCH\( _{3} \)), 55.65 (OCH\( _{3} \)), 55.72 (2OCH\( _{3} \)), 60.21 (OCH\( _{3} \)), 109.68 (C2,6 trimethoxybenzylidene), 111.18 (C2 dimethoxyphenyl), 111.53 (C5 dimethoxyphenyl), 114.70 (C3,5 methoxyphenyl), 120.47 (C1 dimethoxyphenyl), 122.47 (C6 dimethoxyphenyl), 126.15 (C olefinic), 127.60 (C1 trimethoxybenzylidene), 129.42 (C2,6 methoxyphenyl), 129.87 (C1 methoxyphenyl), 137.61 (C4 imidazole), 139.49 (C4 trimethoxybenzylidene), 148.14 (C3 dimethoxyphenyl), 151.66 (C4 dimethoxyphenyl), 152.75 (C3,5 trimethoxybenzylidene), 159.24 (C4 methoxyphenyl), 159.42 (C2 imidazole), 170.26 (C=O imidazole). Anal. Calcd. for C\( _{27} \)H\( _{28} \)N\( _{2} \)O\( _{7} \) (504.53): C, 66.66; H, 5.59; N, 5.55. Found: C, 66.75; H, 5.51; N, 5.65.

3.2.8. General Procedure for the Preparation of (Z)-3-(3,4-Dimethoxyphenyl)-6-oxo-5-(3,4,5-Trimethoxybenzylidene)-5,6-Dihydro-1,2,4-Triazine-2(1H)-Carbothioamide (8)

To a suspension of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene)oxazolone 1 (399 mg, 1 mmol) in glacial acetic acid (20 mL), we added thiosemicarbazide (91 mg, 1 mmol) and anhydrous sodium acetate (98.4 mg, 1.2 mmol). The resulting reaction mixture was heated to for 8 h. After refluxing time, a solid separated out that was filtered, dried and crystallized from ethanol/H\( _{2} \)O (3:1) to furnish pure N-thioamide molecule 8.
3.2.9. General Procedure for the Preparation of (Z)-3-(3,4-Dimethoxyphenyl)-2-Phenyl-5-(3,4,5-Trimethoxybenzylidene)-1,2-Dihydro-1,2,4-Triazin-6(5H)-one (9)

To a suspension of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene)oxazolone 1 (399 mg, 1 mmol) in pure ethanol (20 mL), we added phenyl hydrazine (1 mmol), and the reaction mixture was refluxed for 6 h. After refluxing time, the solvent was evaporated under pressure and the crude solid thus obtained was filtered, washed with water, dried and crystallized from ethanol/H$_2$O (3:1) to afford pure compound 9.

White powder (340 mg, 69.39%), m.p. 248–250 °C. $^1$H-NMR (400 MHz, DMSO-$d_6$, $\delta$ ppm): 3.71 (s, 3H, OCH$_3$), 3.77 (s, 3H, OCH$_3$), 3.81 (s, 3H, OCH$_3$), 3.88 (s, 6H, 2OCH$_3$), 6.71 (d, $J$ = 7.9 Hz, 2H, arom.CH), 6.83 (t, $J$ = 7.3 Hz, 1H, arom.CH), 7.08 (d, $J$ = 8.7 Hz, 1H, arom.CH), 7.15 (s, 1H, olefinic CH), 7.22 (t, $J$ = 7.9 Hz, 2H, arom.CH), 7.85 (d, $J$ = 1.9 Hz, 1H, arom.CH), 7.88 (s, 2H, arom.CH), 7.94 (dd, $J$ = 8.6, 1.9 Hz, 1H, arom.CH), 9.07 (s, 1H, NH). $^{13}$C-NMR (100 MHz, DMSO-$d_6$, $\delta$ ppm): 55.45 (OCH$_3$), 56.16 (3OCH$_3$), 60.69 (OCH$_3$), 110.43 (C2,6 trimethoxybenzylidene), 111.40 (C2 dimethoxyphenyl), 111.94 (C5 dimethoxyphenyl), 112.66 (C2,6 phenyl), 120.13 (C1 trimethoxybenzylidene), 120.51 (C4 phenyl), 122.75 (C6 dimethoxyphenyl), 127.17 (C olefinic), 129.82 (C3,5 phenyl), 130.14 (C1 trimethoxybenzylidene), 135.94 (C5 triazinone), 140.16 (C4 trimethoxybenzylidene), 146.98 (C1 phenyl), 148.94 (C3 dimethoxyphenyl), 152.90 (C4 trimethoxybenzylidene), 159.50 (C3 triazinone), 167.77 (C=O triazinone). Anal. Calcd. for C$_{27}$H$_{27}$N$_3$O$_6$ (489.52): C, 66.25; H, 5.56; N, 8.58. Found: C, 66.32; H, 5.60; N, 8.51.

3.2.10. General Procedure for the Preparation of (Z)-3-(3,4-Dimethoxyphenyl)-2-Isonicotinoyl-5-(3,4,5-Trimethoxybenzylidene)-1,2-Dihydro-1,2,4-Triazin-6(5H)-one (10)

A mixture of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene)oxazolone 1 (399 mg, 1 mmol), isonicotinic acid hydrazide (137 mg, 1 mmol) and sodium acetate anhydrous (98.4 mg, 1.2 mmol) in glacial acetic acid (20 mL) was heated to reflux for 12 h. After refluxing time, the solvent was evaporated under reduced pressure. The crude solid thus obtained was filtered, washed with water, dried and crystallized from DMF/H$_2$O (1:1) to furnish pure compound 10.

White powder (308 mg, 59.42%), m.p. 248–250 °C. $^1$H-NMR (400 MHz, DMSO-$d_6$, $\delta$ ppm): 3.71 (s, 3H, OCH$_3$), 3.77 (s, 3H, OCH$_3$), 3.79 (s, 3H, OCH$_3$), 3.87 (s, 6H, 2OCH$_3$), 7.05 (s, 1H, arom.CH), 7.07 (s, 1H, olefinic CH), 7.84 (s, 2H, arom.CH), 7.87 (d, $J$ = 1.7 Hz, 2H, arom.CH), 7.89 (d, $J$ = 7.5 Hz, 2H, arom.CH), 8.70 (d, $J$ = 5.9 Hz, 2H, arom.CH), 10.03 (s, 1H, NH). $^{13}$C-NMR (100 MHz, DMSO-$d_6$, $\delta$ ppm): 55.12 (OCH$_3$), 55.65 (OCH$_3$), 55.71 (2OCH$_3$), 60.69 (OCH$_3$), 109.68 (C2,6 trimethoxybenzylidene), 110.91 (C2 dimethoxyphenyl), 111.36 (C5 dimethoxyphenyl), 121.33 (C1 triazinone), 121.65 (C3,5 pyridyl), 121.95 (C6 dimethoxyphenyl), 125.30 (C olefinic), 130.05 (C1 trimethoxybenzylidene), 136.79 (C5 triazinone), 139.30 (C4 trimethoxybenzylidene), 148.34 (C3 dimethoxyphenyl), 150.09 (C6 pyridyl), 150.39 (C4 pyridyl), 152.00 (C4 dimethoxyphenyl), 152.74 (C3,5 trimethoxybenzylidene), 160.08 (C3 triazinone), 168.52 (C=O triazinone), 172.36 (C=O isonicotinamide). Anal. Calcd. for C$_{27}$H$_{26}$N$_3$O$_7$ (518.52): C, 62.54; H, 5.05; N, 10.81. Found: C, 62.44; H, 4.99; N, 10.94.

3.2.11. General Procedure for the Preparation of (Z)-3-(3,4-Dimethoxyphenyl)-2-(4-Phenylthiazol-2-yl)-5-(3,4,5-Trimethoxybenzylidene)-1,2-Dihydro-1,2,4-Triazin-6(5H)-one (11)

A suspension of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene)oxazolone 1 (399 mg, 1 mmol) in DMF (20 mL) containing glacial acetic acid (0.5 mL) was added to 2-hydrazinyl-4-phenylthiazole (191 mg, 1 mmol). The reaction mixture was refluxed for 24 h. After cooling, the solution was poured into 100 mL of diethyl ether. The organic layer was washed with water (70 mL) and dried with Na$_2$SO$_4$. The solvent was evaporated
under pressure and the crude product was crystallized from DMF/H$_2$O (1:1) to get pure compound 11.

White powder (293 mg, 51.25%), m.p. 248–250 °C. $^1$H-NMR (400 MHz, DMSO-$_d$$^6$, δ ppm): 3.76 (s, 6H, OCH$_3$), 3.82 (s, 3 H, OCH$_3$), 3.88 (s, 6H, 2OCH$_3$), 7.14 (d, $J$ = 8.6 Hz, 1H, arom.CH), 7.21 (s, 1H, olefinic CH), 7.28 (t, $J$ = 7.1 Hz, 1H, arom.CH), 7.36 (d, $J$ = 7.6 Hz, 2H, arom.CH), 7.39 (s, 1H, thiazole CH), 7.76 (d, $J$ = 7.7 Hz, 2H, arom.CH), 7.79–7.83 (m, 1H, arom.CH), 7.88 (s, 2H, arom.CH), 10.80 (s, 1H, NH). $^{13}$C-NMR (100 MHz, DMSO-$_d$$^6$, δ ppm): 55.13 (OCH$_3$), 55.72 (3OCH$_3$), 60.24 (OCH$_3$), 104.94 (C5 thiazole), 110.09 (C2,6 trimethoxybenzylidene), 110.74 (C2 dimethoxyphenyl), 111.61 (C5 dimethoxyphenyl), 119.39 (C1 dimethoxyphenyl), 121.93 (C6 dimethoxyphenyl), 125.61 (C2,6 phenyl), 127.46 (C4 phenyl), 127.84 (C olefinic), 128.64 (C3,5 phenyl), 129.44 (C1 trimethoxybenzylidene), 134.05 (C1 phenyl), 134.87 (C5 triazinone), 139.88 (C4 trimethoxybenzylidene), 148.62 (C3 dimethoxyphenyl), 150.57 (C4 thiazole), 152.51 (C4 dimethoxyphenyl), 152.78 (C3,5 trimethoxybenzylidene), 157.96 (C3 triazinone), 167.94 (C=O triazinone), 168.49 (C2 thiazole). Anal. Calcd. for C$_{30}$H$_{28}$N$_4$O$_6$S (572.63): C, 62.92; H, 4.93; N, 9.78. Found: C, 63.05; H, 5.02; N, 9.68.

3.3. Biological Evaluation

3.3.1. MTT Cytotoxicity Assay

MTT colorimetric assay was carried out to investigate the impact of the newly synthesized compounds 2–11 on hepatocellular carcinoma (HepG2) and normal liver cell line (HL-7702) cell lines. See Section 3.3.1 in Supplementary Material.

3.3.2. Tubulin Polymerization Assay

To explore the cytotoxic activity of the prepared compounds, tubulin polymerization assay of the most potent compounds 9, 10 and 11 was performed according to the reported method [2]. See Section 3.3.2 in Supplementary Material.

3.3.3. Cell Cycle Analysis

Cell cycle analysis in HepG2 cells was carried out by FACS analysis according to the manufacturer’s directions. See Section 3.3.3 in Supplementary Material.

3.3.4. Annexin V/FITC Staining Assay

Annexin V/FITC double staining analysis in HepG2 cells was carried out by FACS analysis according to the manufacturer’s directions. See Section 3.3.4 in Supplementary Material.

3.3.5. Mitochondrial Membrane Potential

MMP was measured by flow cytometry analysis in HepG2 cells according to the manufacturer’s directions. See Section 3.3.5 in Supplementary Material.

3.3.6. Effect on p53, Bax and Bcl-2

Apoptotic markers (p53, Bax and Bcl-2) were measured in HepG2 cells using ELISA analysis according to the manufacturer’s directions. See Section 3.3.6 in Supplementary Material.

3.3.7. Molecular Docking Study

All the docking study was performed using molecular operating environment (MOE 2009.10). See Section 3.3.7 in Supplementary Material.

4. Conclusions

In conclusion, a series of novel trimethoxyphenyl-based analogs were synthesized by varying the azalactone ring of 2-(3,4-dimethoxyphenyl)-4-(3,4,5-trimethoxybenzylidene) oxazolone 1, and they were tested for their cytotoxic activity against the hepatocellular carcinoma HepG2 cell line and the normal liver HL-7702 cell line. The tested molecules exhibited good activity over the HepG2 cells, especially compounds 9, 10, and 11 with
IC_{50} values of 1.38 ± 0.15, 2.52 ± 0.24, and 3.21 ± 0.22 μM, respectively, compared to podo (IC_{50} = 2.08 ± 0.04 μM). In addition, compounds 9, 10, and 11 exhibited β-tubulin polymerization inhibition percentage well, with percentage inhibitions of 86.73%, 80.51%, and 69.51%. Additionally, cell cycle analysis of compound 9 revealed cell cycle disturbance at the G2/M phase (3-fold more than control HepG2). Furthermore, the Annexin V-FITC/PI dual staining assay results show that treatment of HepG2 cells with compound 9 induced a significant increase in Annexin-V-positive cells by almost 18-fold compared with the untreated control. Moreover, Compound 9 was further studied regarding its apoptotic potential in HepG2 cells; it decreased the level of MMP and Bcl-2 by 3.12- and 5.58-fold, respectively, compared with control HepG2 cells. It also boosted the level of p53 and Bax by 8.91- and 4.50-fold, respectively, compared with control HepG2 cells.

**Supplementary Materials:** The following supporting information can be downloaded at: [https://www.mdpi.com/article/10.3390/molecules27144621/s1](https://www.mdpi.com/article/10.3390/molecules27144621/s1), Figure S1: 1H-NMR spectrum of compound 1, Figure S2: 13C-NMR spectrum of compound 1, Figure S3: 1H-NMR spectrum of compound 2a, Figure S4: 13C-NMR spectrum of compound 2a, Figure S5: 1H-NMR spectrum of compound 2b, Figure S6: 13C-NMR spectrum of compound 2b, Figure S7: 1H-NMR spectrum of compound 3a, Figure S8: 13C-NMR spectrum of compound 3a, Figure S9: 1H-NMR spectrum of compound 3a, Figure S10: 13C-NMR spectrum of compound 3b, Figure S11: 1H-NMR spectrum of compound 4a, Figure S12: 13C-NMR spectrum of compound 4a, Figure S13: 1H-NMR spectrum of compound 4b, Figure S14: 13C-NMR spectrum of compound 4b, Figure S15: 1H-NMR spectrum of compound 4c, Figure S16: 13C-NMR spectrum of compound 4c, Figure S17: 1H-NMR spectrum of compound 5, Figure S18: 13C-NMR spectrum of compound 5, Figure S19: 1H-NMR spectrum of compound 6, Figure S20: 13C-NMR spectrum of compound 6, Figure S21: 1H-NMR spectrum of compound 7a, Figure S22: 13C-NMR spectrum of compound 7a, Figure S23: 1H-NMR spectrum of compound 7b, Figure S24: 13C-NMR spectrum of compound 7b, Figure S25: 1H-NMR spectrum of compound 7b, Figure S26: 13C-NMR spectrum of compound 8, Figure S27: 1H-NMR spectrum of compound 9, Figure S28: 13C-NMR spectrum of compound 9, Figure S29: 1H-NMR spectrum of compound 10, Figure S30: 13C-NMR spectrum of compound 10, Figure S31: 1H-NMR spectrum of compound 11, Figure S32: 13C-NMR spectrum of compound 11.

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