Volatile Fatty Acids from Lipid-Extracted Yeast Provide Additional Feedstock for Microbial Lipid Production

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Abstract: Microbial lipid production from oleaginous yeasts is a promising process for the sustainable development of the microbial biodiesel industry. However, the feedstock cost poses an economic problem for the production of microbial biodiesel. After lipid extraction, yeast biomass can be used as an organic source for microbial biodiesel production. In this study, volatile fatty acids (VFAs), produced via anaerobic digestion of a lipid-extracted yeast (LEY) residue, were utilized as a carbon source for the yeast Cryptococcus curvatus. The response surface methodology was used to determine the initial pH and inoculum volume for the optimal VFA production. The experimental result for VFA concentration was 4.51 g/L at an initial pH of 9 and an inoculation 25%. The optimization results from the response surface methodology showed that the maximal VFA concentration was 4.58 g/L at an initial pH of 8.40 and an inoculation of 39.49%. This study indicates that VFAs from LEY can be used as a carbon source for microbial biodiesel production, with the potential to significantly reduce feedstock costs.

Keywords: oleaginous yeast; volatile fatty acid; lipid-extracted yeast; response surface methodology; Cryptococcus curvatus

1. Introduction

The prospect of converting microbial lipids into biodiesel has attracted attention from many researchers because of the renewable potential of biofuel sources such as microalgae, fungi, and yeasts. Oleaginous yeasts are characterized by remarkable growth and show promise as a biofuel feedstock because of their high oil content (up to 67.5%) [1–4]. However, biodiesel production from microbial lipids is much more expensive than from fossil fuels, because of the high costs of raw materials. The commodity feedstock cost accounts for the largest proportion of the total biofuel cost [5]. Considering a microalgae lipid content of 50%, the per-liter production cost of biodiesel from algal oil varies from USD 1.08 to USD 1.34 [6,7]. Furthermore, yeasts such as Cryptococcus albidus [7–9], Cryptococcus curvatus [10–12], Yarrowia lipolytica [13,14], and Rhodosporidium toruloides [15,16], which can use low-cost volatile fatty acid (VFAs) materials (USD 20–100/ton) as carbon sources, can produce biodiesel at a cost of $0.3–$1.26/L [7]. VFAs are mainly composed of acetic acid, propionic acid, and butyric acid and are produced anaerobiacally from a variety of biomasses, including organic waste biomass, sludge, animal manure, microbe, plant, and animal biomass sources [7,17,18]. Hence, reducing the raw material-associated costs can economically impact biofuel production, with carbon source recycling serving as one of the major cost-reduction approaches. In addition, the maximum theoretical yield of microbial lipid yield is quite low (around 0.32 g/g from glucose) and one-third carbon...
is released by CO2 producing Acetyl-CoA, which is the building block of fatty acids, during glucose metabolism [1]. Hence, it is necessary to increase carbon efficiency and lipid yield. A substantial amount of oleaginous yeast residue is generated after lipid extraction. Production of VFAs via anaerobic digestion of a lipid-extracted yeast (LEY) residue can potentially provide an additional carbon source. The recycling of VFAs via their utilization as carbon sources for oleaginous yeasts can lower biofuel production costs [7,9,19]. Although C. curvatus is a well-known high lipid-producing yeast, there are only a few studies on the production of VFAs from LEY. Therefore, in this study, a lipid-extracted C. curvatus cell residue was evaluated as a source for VFA production via anaerobic digestion by varying two parameters, the pH and the inoculation [18,20–22].

2. Results and Discussion

2.1. Preparation of a LEY Residue

The cell mass and lipid content were 12.08 g/L and 49.3%, respectively, after 72 h of the cultivation of C. curvatus (Figure S1). The cell wall of the ellipsoid yeast cells (Figure 1A) was disrupted (Figure 1B) using an ultrasonic homogenizer in Folch solvent. After lipid extraction, the lipid content changed from 49.3% to 13.0% in the cell residue.

![Scanning electron microscopy images of C. curvatus cells. (A) Before lipid extraction (5000×) and (B) after lipid extraction (10,000×).](image)

2.2. Application of Response Surface Methodology for the Optimization of VFA Production via Anaerobic Digestion

The data from 11 trials conducted to assess VFA concentration responses using the CCD method are presented in Table 1. The experiments using the conditions represented by the center point, initially designated at pH 9.0 and an inoculation ratio of 25%, were repeated three times to estimate experimental errors.

| Run | pH | Inoculation (% v/v) | VFA Concentration (g/L) |
|-----|----|---------------------|-------------------------|
| 1   | 6.5| 5                   | 3.10                    |
| 2   | 6.5| 25                  | 3.75                    |
| 3   | 6.5| 45                  | 3.41                    |
| 4   | 9.0| 5                   | 3.82                    |
| 5*  | 9.0| 25                  | 4.20                    |
| 6*  | 9.0| 25                  | 4.51                    |
| 7*  | 9.0| 25                  | 4.27                    |
| 8   | 9.0| 45                  | 4.68                    |
| 9   | 11.5| 5                   | 1.38                    |
The independent variables, the initial pH (X₁) and the inoculation (X₂), were linked to the VFA concentration by quadratic Equation (1):

\[ \text{VFA concentration} = 4.37 - 0.86 X₁ + 0.29 X₂ + 0.067 X₁X₂ - 1.69 X₁^2 - 0.19 X₂^2 \]  

The three-dimensional response surface plot of the quadratic model for VFA concentrations, with putative estimated optimums, is displayed in Figure 2. According to the response surface, the optimum condition fell within the boundaries of the design. Regression models that described the experimental data indicated correlations among the two independent parameters that affected the response, with an \( R^2 \) value of 0.97 for the VFA concentrations. This \( R^2 \) value highlights the adequate process model representation, as well as a good fit between the predicted and experimental results.

Table 2 shows the quadratic model analysis of variance (ANOVA) data for optimizing the VFA concentration during acidogenesis, involving two independent variables, as well as their interactions. Corresponding coefficients, with the statistical regression coefficients, were tested. The model confirmed using the given response surface was considered not significant with a lack of fit, but significant with an \( \alpha \)-level regression of 1%. According to the significance determination of the independent variables, inoculation exhibited a trivial impact on the VFA concentration compared to initial pH from the ANOVA results. The maximum condition of VFA calculations, previously defined as the optimum conditions, were obtained by considering the partial derivatives as zero, based on the corresponding variables. The maximum VFA concentration was 4.58 g/L at an initial pH of 8.40 and inoculation of 39.49%. The estimates revealed that we could fit a quadratic model having an identical design boundary to the general environment of the VFA response surface. The model also exhibited a response plot that proceeded diagonally from the lower to the upper region with rounded ridges.

Table 2. Analysis of variance of the experimental VFA concentration data.

| Source   | Degrees of Freedom | VFA Concentration | Mean Square | p-Value |
|----------|--------------------|-------------------|-------------|---------|
| Model    | 5                  |                   | 2.67        | 0.0001  |
| X₁       | 1                  |                   | 4.45        | 0.0001  |
| X₂       | 1                  |                   | 0.51        | 0.0151  |
| X₁X₂     | 1                  |                   | 0.018       | 0.5243  |

* Center point was repeated three times.
\begin{tabular}{lccc}
\hline
 \textbf{Term} & \textbf{DF} & \textbf{SSE} & \textbf{MS} \\
\hline
$X_1^2$ & 1 & 7.26 & <0.0001 \\
$X_2^2$ & 1 & 0.091 & 0.1863 \\
Residual & 5 & 0.039 & \\
Lack of fit & 3 & 0.047 & 0.3782 \\
Pure error & 2 & 0.026 & \\
Correlation total & 10 & & \\
\hline
\end{tabular}

\[ R^2 = 97.15\% \text{ for VFA concentration} \]

Defined response surfaces of the optimum response for unbounded polynomial models could be beyond the design boundary. These types of models extend the polynomial models to infinity. Hence, when predicting response surfaces, the models should include specific boundaries.

To validate the model, a point with an initial pH of 8.0 and inoculation of 10% was randomly selected for the experiment. In this trial, the experimental VFA concentration was 3.89 g/L, while the concentration after substitution in Equation (1) was 4.14 g/L. The error between the model and experimental data was <10%, confirming that the model equation could correctly fit the experimental data.

2.3. Recycling of the LEY Residue as a Feedstock for VFA Production

The economic feasibility of industrial-scale biodiesel production is limited by the cost of raw materials, which account for more than 50% of the total production costs [23]. In our study, the LEY residue was used to produce VFAs by anaerobic digestion, and the yield was 0.458 g VFAs/g LEY. Furthermore, \textit{C. curvatus} could efficiently utilize VFAs as a carbon source, yielding 0.62 g cells/g VFAs and 0.13 g lipids/g VFAs from our previous study [12]. Hence, 130 g of lipids was produced from 1 kg of VFAs by the oleaginous yeast (Figure 3A). Concurrently, 490 g of the cell residue (cell mass—lipids) was converted into VFAs via anaerobic digestion, yielding 224.4 g of VFAs (Figure 3B). Considering VFAs as an additional carbon source for yeast cultivation, the lipid production was increased by 22% through recycling (Figure 3C). Furthermore, if VFAs from LEY are considered as a carbon source for microbial lipid production, with persistent recycling, the total VFA production is calculated using an infinite geometric series, expressed as follows:

\[ \sum_{k=1}^{\infty} a r^{k-1} = \frac{a}{1-r}, \]

where \( k \) is the recycling number, \( a \) denotes the initial VFA mass, and \( r \) represents the geometric ratio.
Figure 3. Mass balance of microbial lipid production from volatile fatty acids (VFAs) as low-cost materials and recycling of lipid-extracted yeast (LEY) residues. (A) Microbial lipid production from VFAs. (B) Microbial lipid production from VFAs and VFA production from the LEY residue. (C) Microbial lipid production from VFAs and additional VFAs by the one-time recycling of the LEY residue. (D) Microbial lipid production from VFAs and additional VFAs by the unlimited recycling of LEY residues.

In the present study, the geometric ratio was calculated as follows:

\[
\text{Cell residue yield from VFAs} \times \text{VFA yield from the residue} = 0.490 \times 0.458 = 0.2244 \quad (3)
\]

Based on Equations (2) and (3), unlimited recycling would increase the VFA concentration by 28.9% relative to that without recycling (Figure 3D). According to the mass balance of lipid production, 1 kg of the oleaginous yeast accounted for 493 g of lipids and 507 g of the cell residue. Hence, 299.4 g of VFAs could be produced from the recycling of the cell residue, with 38.9 g of converted lipids. Therefore, 1 kg of the oleaginous yeast would produce a maximum of 531.9 g of lipids. The yield of VFAs from LEY depends on the lipid extraction efficiency. However, we assumed that the lipid extraction efficiency was 100% and used the same yield data (0.458 g/g) for the convenience of calculation in Figure 3.
3. Materials and Methods

3.1. Yeast Cultivation

*C. curvatus* (ATCC 20509, KCTC 7225) was obtained from the Korea Biological Resource Center (Daejeon, Republic of Korea). The strain was maintained by a monthly subculture on potato dextrose agar (2% dextrose, 0.4% potato infusion, and 1.5% agar). Seed culture was precultured for 25 h on YPD (2% dextrose, 2% peptone, 1% yeast extract) at 25 °C. This culture was used for batch culture as an inoculum, and 10% (v/v) inoculation was used. The batch culture was performed using a minimal medium (pH 5.5) in a 2.5 L fermenter containing the following per liter of distilled water: Glucose 20 g, NaNO₃ 1.2 g, KH₂PO₄ 2.7 g, Na₂HPO₄ 0.95 g, MgSO₄·7H₂O 0.2 g, yeast extract 0.1 g, and EDTA 0.1 g. Supplemented with a 100× stock solution, which contained per liter: CaCl₂·2H₂O 4 g, FeSO₄·7H₂O 0.55 g, citric acid H₂O 0.52 g, ZnSO₄·7H₂O 0.10 g, MnSO₄·H₂O 0.076 g, and 100 μL 18M H₂SO₄ [12,24]. Dry cell mass was determined by harvesting and transferring the culture broth to a pre-weighed tube after centrifugation. After washing twice with distilled water, the samples were dried at 85 °C overnight to achieve a constant weight.

3.2. Lipid Extraction of Yeast and Preparation LEY Residue Sample

Lipid extraction was done with the Folch method [25] with minor modifications: 1.5 g of dry biomass (after lyophilization) was mixed with 30 mL Folch solvent (2:1 v/v of chloroform and methanol), with sonication for 30 min to break the cell wall. The mixture was then centrifuged at 8000 rpm for 10 min and solvent discarded. Then, we washed twice with methanol to remove the remaining chloroform, washed with distilled water twice, and then lyophilized the sample.

3.3. Anaerobic Digestion Operating Conditions

The inoculum used was acquired from the Daejeon sewage disposal plant (total solids = 10.57 g/L, volatile solids = 4.45 g/L) and stored at 4 °C. Then, 10 g/L of yeast residue and different inoculation was used for each experiment. Modified RAMM medium (1 g/L yeast extract, 0.27 g/L KH₂PO₄, 0.35 g/L K₂HPO₄, 0.53 g/L NH₄Cl, 0.1 g/L MgCl·6H₂O, 0.075 g/L CaCl₂·2H₂O) was used, with 5 g/L NaHCO₃ by alkaline buffer. Iodoform (15 ppm) was used as a methane inhibitor [18,21]. To the fermentation medium, 50 μL of both DSMZ 320 trace element solution and DSMZ medium 503 vitamin solution was added. The total volume of each culture was brought to 100 mL in a 250 mL flask. Fermentation was conducted in a shaking incubator with 250 mL flasks that were sealed with silicon rubber stoppers, after which anaerobic digestion proceeded for 2 weeks.

3.4. VFA Concentration Analysis

VFA samples from the anaerobic digestion were analyzed by HPLC using an Aminex HPX-87H column (Bio-Rad Laboratories, Hercules, CA, USA) with a refractive index (RI) detector. Mobile phase was 5 mM of H₂SO₄ at a 0.6 mL/min flow rate and temperature was 65 °C. Before HPLC analysis, all samples were filtered through a 0.22 μm PVDF filter.

3.5. Central Composite in Cube in Design and Selection of Variables

A central composite (CCD) design for two independent variables was used for approaching the optimum settings. The CCD design, which consists of 2² factorial design augmented by a center (n = 3), along with 2 × 2 axial points, was employed in this study with a slight modification. The concentrations of VFAs, including acetic, propionic, and butyric acids were determined at each CCD point. As shown in Table 3, pH (X₁) and inoculation (X₂) were chosen as the two independent factors in the experimental design. The central values of the experiment design were selected as pH 9 and 25% (v/v) inoculation at mesophilic (35 °C) temperature. Regression analyses were performed using experimental data. To optimize and analyze the factors that affected the total concentration of
VFAs, including acetic, propionic, and butyric acid, in the acidogenic stage, RSM was applied. A test of the validity of the model was used to estimate polynomials (Equation (4)) [26].

\[
\eta_a = c_0 + \sum_{i=1}^{n} \alpha_i x_i + \sum_{i=1}^{n} \alpha_i x_i^2 + \sum_{i} \sum_{j} \alpha_i x_i x_j \tag{4}
\]

where \(\eta_a\): the experimental value of \(a\) (\(a = \text{VFAs concentration}\)); \(x_i\): independent variable \(k\) (\(k = \text{initial pH and inoculation in that order}\)); \(c_0\): regression constant; \(\alpha_i\): regression coefficients of the independent variable \(k\) (\(k = \text{initial pH and inoculation in that order}\)).

Table 3. Data for independent variables and their levels used in the experimental design.

| Factor          | Coded Levels | -1   | 0   | 1   |
|-----------------|--------------|------|-----|-----|
| pH, X1          |              | 6.5  | 9.0 | 11.5|
| Inoculation (%) |              | 5    | 25  | 45  |

The least-squares method was used to estimate the parameters in the approximating polynomials. A pH range between 6.5 and 8.5 was selected for the first experiment. Subsequently, the optimum point was determined to be a boundary condition; therefore, the pH condition was changed from 6.5 to 11.5. (Table 3). Eleven trials were performed to approximate the response.

4. Conclusions

Bioenergy and biofuel are advantageous for reducing greenhouse gas emissions and carbon neutrality. However, their widespread utilization is hindered by economical limitations, particularly the feedstock cost. We studied microbial lipids suitable for conversion into biodiesel and additional lipid production from yeast-residue VFAs after lipid extraction. The VFA production was optimized using an inoculation ratio of 39.49% and an initial pH of 8.4, which yielded 0.458 g VFAs/g LEY residue. Cell residue recycling increased the lipid yield from the substrate by 28.9%. However, additional process costs include extraction, anaerobic digestion, and yeast cultivation during recycling. Hence, microbial lipid production from LEY increases the total lipid production, but recycling requires additional process costs. Therefore, the LEY recycling frequency should be carefully considered to make the approach feasible.

Supplementary Materials: The following are available online at www.mdpi.com/article/10.3390/catal11081009/s1, Figure S1: Growth curve and accumulation of lipids during incubation of *C. curvatus* for 72 h.

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