Quantum Defects as a Toolbox for the Covalent Functionalization of Carbon Nanotubes with Peptides and Proteins

Florian A. Mann, Niklas Herrmann, Felipe Opazo, and Sebastian Kruss*

Abstract: Single-walled carbon nanotubes (SWCNTs) are a 1D nanomaterial that shows fluorescence in the near-infrared (NIR, >800 nm). In the past, covalent chemistry was less explored to functionalize SWCNTs as it impairs NIR emission. However, certain sp³ defects (quantum defects) in the carbon lattice have emerged that preserve NIR fluorescence and even introduce a new, red-shifted emission peak. Here, we report on quantum defects, introduced using light-driven diazonium chemistry, that serve as anchor points for peptides and proteins. We show that maleimide anchors allow conjugation of cysteine-containing proteins such as a GFP-binding nanobody. In addition, an Fmoc-protected phenylalanine defect serves as a starting point for conjugation of visible fluorophores to create multicolor SWCNTs and in situ peptide synthesis directly on the nanotube. Therefore, these quantum defects are a versatile platform to tailor both the nanotube’s photophysical properties as well as their surface chemistry.

Introduction

Since their discovery, single-walled carbon nanotubes (SWCNTs) attracted a lot of attention not only because of their unique chemical structure, but also because of their outstanding photophysical properties such as non-bleaching/blinking near-infrared (NIR) fluorescence.[1-4] This NIR fluorescence is beneficial especially for bioimaging as the emission wavelength of SWCNTs falls into the so-called tissue-transparency window where the absorption of water, hemoglobin, and lipids reaches a combined minimum and scattering is reduced compared to that of visible light.[5] Consequently, SWCNTs already found application in diverse settings ranging from in vivo NIR imaging[6-8] over drug delivery vehicles[9,10] to NIR optical sensors.[11-20] In order to carry out their desired function in these important applications, the pure carbon tube has to be modified with, for example, the cargo to be carried or a recognition unit imparting specificity for the analyte to be detected. Furthermore, carbon nanotubes are not water-soluble, preventing applications in aqueous systems unless the hydrophobic surface is coated with an amphiphilic surfactant.[12] In the last 20 years, both covalent and noncovalent functionalization have been employed for the decoration of SWCNTs with functional units. Noncovalent functionalization such as adsorption of DNA is by far the most frequently applied approach owing to its ease-of-use and mild conditions.[21-23] Furthermore, conformational changes of the coating molecule can directly translate into changes of the NIR fluorescence, which is interesting for sensing.[16,24] On the other hand, covalent functionalization leads to more stable conjugates, but destroys the SWCNT’s extended π-network and thus also the NIR fluorescence.[25,26] In 2017, Setaro et al. reported preserved fluorescence by using a [2+1]-cycloaddition with electron-poor aromatic nitrenes, which they also used for the attachment of gold nanoparticles and spiropyranes.[26] Very recently, Godin et al. used spiropyran-switchable SWCNTs for NIR superresolution microscopy,[27] while Chio et al. used the same nitrene [2+1]-cycloaddition for the attachment of small molecules such as biotin.[28]

However, sp³ defects can not only diminish the NIR fluorescence, but were also found to modulate it depending on the nature and density of the defects. In 2010, Ghosh et al. reported on a NIR emission peak red-shifted by approximately 130 nm (termed E₁₁) upon introduction of oxygen defects.[29] Later, Piao et al. observed a similar peak shift and enhanced fluorescence quantum yield upon introduction of aryl defects using diazonium salts (quantum defects).[30] This technique also enabled tuning of the defect fluorescence both in terms of intensity and emission wavelength via different substituents on the aryl/alkyl defect.[31-34]

In the last years, a number of different oxygen and aryl defects were reported that are now very promising tools for the generation of brighter/modified SWCNTs[31,35,36] pH[37] and saccharide[38] sensors, short fluorescent SWCNTs for superresolution microscopy,[39] and single-photon sources for quantum computing.[40,41] Furthermore, this new defect-induced fluorescence feature moves the emission even further into the biotransparency window, leading to even better tissue penetration properties.[42,43]

Results and Discussion

In this work, we expand the use of quantum defects from modulation of the SWCNTs’ photophysical properties to-
wards using them as modular anchors for the attachment of peptides and proteins (Figure 1). To this end, we employed an (N-maleimido)phenyl diazonium salt (MalPh-Dz, 2), which—after defect introduction—can be targeted by thiol-containing molecules such as proteins, while at the same time generating a further red-shifted E_{11}’ emission feature (defect-carrying SWCNTs are thus referred to as SWCNT* in the following).

We started our investigations by optimizing the narrow window of reaction conditions[44] between sodium dodecylbenzenesulfonate (SDBS)-dispersed SWCNTs 1 and 4-(N-maleimido)phenyl diazonium tetrafluoroborate (MalPh-Dz, 2), allowing for the NIR fluorescence to be altered but not diminished (Figure 2a). For a fast and efficient screening of conditions (reaction time and reactant concentration, see Figure S1), we made use of a 96-well green LED array for SWCNT excitation driving the radical arylation reaction. The best results were achieved when a 10 nM SWCNT (c_{carbon} ≈ 530 μm) solution (dispersed in 1% SDBS/H_{2}O) was mixed with 2 (100 μm) and irradiated for 15 minutes. Higher MalPh-Dz concentrations would lead to too many defects and diminish NIR fluorescence.[30]

The success of the defect reaction was monitored by NIR fluorescence spectroscopy (Figure 2b) and the SWCNTs’ structural integrity and colloidal stability was controlled using Vis/NIR absorbance spectroscopy (Figure S2). The fluorescence spectra and the excitation/emission map (see Figure 2c) clearly show a growing E_{11}’ peak at approximately 1135 nm with increasing concentration of 2 (control 2D spectrum with a nonmodified SWCNT/SDBS sample can be found in Figure S3). The increased E_{11}/E_{1} ratio of the 2D versus the 1D NIR fluorescence spectrum could be attributed to prolonged exposure to MalPh-Dz 2 before spin filtration and measurement of the 2D spectrum or due to preferential redispersion of MalPh-SWCNT*.

After successful introduction of (N-maleimido)phenyl quantum defects (MalPh defects), we tested whether biomolecules can be conjugated to this anchor. As a first example, we chose a nanobody against green fluorescent protein (GFP). Nanobodies are the isolated antigen-binding region of heavy-chain antibodies found, for example, in Camelidae and are only 10% of the size of conventional antibodies.[45–46] This renders them very useful as binders for diverse applications such as superresolution microscopy,[47–49] live-cell immunostaining after modification with cell-penetrating peptides,[50] and isotopic labeling of biological samples for secondary-ion mass spectrometry (SIMS) imaging.[51] Nanobodies binding GFP (GFP-binding protein, GBP) in particular can be used as a platform technology due to the widespread availability of GFP-fusion proteins or even whole genetically modified organisms expressing GFP-fusion proteins, giving the possibility to target a whole variety of proteins with just one single conjugate. Similar to our previous (noncovalent) work,[52] we used a GBP with a single ectopic C-terminal cysteine for oriented conjugation to the MalPh-SWCNT* 3 leaving the antigen-binding region pointing away from the SWCNT* surface.

With the fast hydrolysis kinetics of N-aryl maleimides,[52] in mind, we evaluated both the sequential defect introduction followed by nanobody conjugation as well as a one-step approach combining all three reaction partners at once (Figure 3a). For the sequential reaction, the excess diazonium salt 2 was removed using 300 kDa molecular weight cutoff (MWCO) spin filters followed by resuspension of the now naked MalPh-SWCNT* 3 in 1x phosphate-buffered saline (PBS, pH 7.4) and reaction with 500 equiv. (≈ 25 equiv. with respect to introduced maleimides) of the nanobody 4 (16 h at room temperature). In the one-step approach, the same excess of GBP was added directly during the defect introduction (30 minutes instead of 15 minutes for the sequential reaction) and left to react for 16 h at room temperature as well. After defect-introduction/bioconjugation, the excess nanobody was removed using 300 kDa-MWCO-spin-filtration and the SWCNT*-GBP conjugate 5 resuspended in 1xPBS using 1.2-distearoyl-sn-glycero-3-phosphoethanolamine-N-[methoxy(polyethylene glycol)-5000] (ammonium salt, PL-PEG5000). As shown in Figure S4, the redispersion step was efficient only for the conjugate 5 synthesized via the one-step approach, yielding a highly concentrated solution (OD = 1.8). Consequently, we proceeded with the resuspended conjugate 5 resulting from the one-step procedure in the following.

Figure 3b shows the NIR fluorescence spectra of the conjugate 5 as well as the negative control, where no diazonium salt 2 was added, both resuspended using PL-PEG5000. A comparison of both spectra shows successful introduction of sp² quantum defects and E_{11}’ emission. To further evaluate whether these defects also contain the covalently attached nanobody, we employed atomic force microscopy (AFM). Here, the one-step-synthesized sample clearly shows nanobodies attached to the SWCNT* (see Figure 3c, control without 2 in Figure S5) with the additional height introduced by the GBP fitting both the value measured by AFM (d = 4.3 ± 0.9 nm) as well as the diameter obtained from the crystal structure (PDB: 3G9A, d = 3.4–4.3 nm, see Figure S6). In a few cases, also larger heights of SWCNT*-

---

**Figure 1.** Defects as a generic handle to functionalize SWCNTs. Differently substituted aryl defects are introduced into the nanotube’s sidewall and used to conjugate biomolecules such as nanobodies (1) or for the direct growth of peptide chains on the SWCNT* (2). Fmoc = fluorenlymethoxycarbonyl.
GBP conjugates (approximately 7 nm) were measured, which could be explained by the possible side reaction of the diazonium salt with the GBP aromatic residues leading to dimer formation. Taken together, these results indicate successful conjugation of the nanobody to maleimide-bearing quantum defects, which in turn are still able to modulate the SWCNT’s NIR fluorescence, yielding emission at 1143 nm.

As a next step, we verified that the nanobody is still able to bind GFP even after covalent conjugation to the SWCNT. After having successfully established MalPh quantum defects as an anchor for the attachment of (immuno)proteins, we wanted to challenge this defect-based approach even by blocking (with bovine serum albumin [BSA])/washing steps and incubation with the conjugate 5.

The observed colocalization of the GFP and the NIR channel indicates retained function of the GBP conjugate even after covalent conjugation to the SWCNT. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentration) were chosen to immobilize only a few GBPs per tube. The reaction conditions (low defect density, reactant concentra...
research articles

Further with the aim of synthesizing peptide chains directly on the SWCNT sidewall. While there are a few reports on the use of coiled-coil or cyclic peptides for SWCNT dispersion and the noncovalent immobilization of RGD motifs is less explored. Pantarotto et al. and Bianco et al. utilized the 1,3-dipolar cycloaddition of azomethine ylides for the covalent modification of SWCNTs and for subsequent attachment of previously synthesized, short peptides. However, this approach destroys the SWCNT’s optical properties and rules out NIR fluorescence imaging applications.

To use quantum defects as a starting point for peptide growth, we synthesized a diazonium salt containing a fluorenylmethoxycarbonyl (Fmoc)-protected l-phenylalanine (Fmoc-Phe-Dz, 6) in a one-step procedure. Next, we again optimized the conditions for defect introduction (Fmoc-Phe defects) and evaluated the success using 1D/2D NIR fluorescence spectroscopy (Figure 4b,c, control 2D spectrum in Figure S8). We carried out the Fmoc deprotection using 20% piperidine/DMF in a 1 mL syringe reactor equipped with a standard 20 μm pore-size frit. After washing, we coupled the fluorophore 5(6)-carboxyfluorescein (CF) to assess addressability of the unprotected amine. Figure 4d shows colocalization of the NIR and the CF channels after immobilization on glass and washing steps using 1xPBS with 0.1 % Triton-X-100 as opposed to the negative controls [without 6 (Figure 4d) or without Fmoc deprotection (Figure S9)]. This result shows that the unprotected amine is still addressable and the conjugation of carboxyfluorescein led to the generation of covalently linked multicolor SWCNTs*. Encouraged by these promising results, we wanted to test next whether it is also possible to synthesize a whole peptide sequence on the Fmoc-Phe-SWCNT* 7. Here, we chose a positively charged hexaarginine peptide to also evaluate its impact on SWCNT* solubility in aqueous environments (Figure 5a). To evaluate the success of the SWCNT*-peptide synthesis, neither the Kaiser test for free amines nor UV measurements after Fmoc cleavage could be used due to their insufficient sensitivities on the small scale of these experiments (η0(SWCNT) = 100 pmol in 1 mL solution). Thus, we decided to couple 5(6)-CF to the N-terminus before global deprotection of the arginine’s side chains using a deprotection cocktail [75 % trifluoroacetic acid (TFA), 15 % dichloromethane (DCM), 5 % ddH2O, 5 % trisopropylsilane (TIS)]. This was followed by tip-sonication in 1xPBS (3 min, 30 % amp, 4°C) and centrifugation (16 100 g, 30 min) to remove insoluble SWCNTs. Figure 5b shows colocalization of the NIR and the CF channels on the single-nanotube level, indicating successful synthesis of SWCNT*-F-R_C in contrast to the control (without 6, see Figure S10) and retained optoelectronic properties after TFA deprotection. In fact, the negative control did not contain any SWCNTs, indicating increased solubility in aqueous environments by covalent peptide functionalization. However, due to the small number of defects (approximately one defect per 20 nm tube), the SWCNT*-F-R_C (8) did not display high solubility in water and therefore additional wrapping was used to increase the concentration and colloidal stability. In future studies, this aspect could be further evaluated with higher defect densities and/or longer peptide sequences.

As a next step we scaled up the SWCNT*-based peptide synthesis and synthesized multiple SWCNT*-peptide conjugates at the same time in a 96-well format (equipped with 0.2 μm pore size filters, Figure 5c). Again, the synthesis followed the same protocol as above, yet with smaller reaction volumes. Using this technique, we synthesized twelve different peptide sequences directly on NIR fluorescent carbon nanotubes. After side-chain deprotection using the same deprotection cocktail as for SWCNT*-F-R_C, the carbon nanotubes were redispersed in an aqueous 1 % DOC solution via tip-sonication. While DOC leads to slightly red-shifted emission compared to SDBS-dispersed SWCNT*-Fmoc/ SDBS (7) both Figure 5d and e show the impact of peptide sequence on the NIR fluorescence. For some sequences the original E11 peak was almost twice as intense as the E11 peak—other sequences showed exactly the

Figure 4. Fmoc-phenylalanine quantum defects and multicolor SWCNTs. a) Strategy for defect introduction and subsequent Fmoc deprotection followed by CF conjugation. DIPEA = N,N-diisopropylethylamine, HATU = 1-[bis(dimethylamino)methylene]-1H-1,2,3-triazolo[4,5-b]pyridinium 1-oxide hexafluorophosphate. b) NIR fluorescence spectra of SWCNTs treated with different concentrations of Fmoc-Phe-Dz showing increased E11/E11* ratios at higher diazonium salt concentrations. c) Excitation-emission map of Fmoc-Phe-SWCNT* showing the E11* fluorescence amongst other minor SWCNT species and E11 fluorescence. d) SWCNT*-Phe-CF immobilized on a glass slide showing colocalization of the NIR (≥ 900 nm) and the CF channels (500–550 nm), whereas the control without sp2 defects does not show a CF signal, indicating successful conjugation. Scale bars = 5 μm.

www.angewandte.org 17735

Angew. Chem. Int. Ed. 2020, 59, 17732 – 17738 © 2020 The Authors. Published by Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim
opposite behavior with the \( E_{11}^* \) signal being 2.5-fold stronger than the signal arising from the \( E_{11} \) transition (Vis/NIR absorbance and NIR fluorescence spectra in Figure S11). A closer evaluation of the sequence dependence of this fluorescence modulation shows that the \( E_{11}^*/E_{11} \) ratio decreases with an increasing number of hydrophobic residues in the peptide sequence attached to the defect responsible for exciton trapping (Figure 5f). A similar effect was already observed by Kwon et al., who found changing \( E_{11}^*/E_{11} \) emission wavelengths and \( E_{11}^*/E_{11} \) ratios for differently substituted (fluoro)alkyl/aryl sp\(^3\) defects.\[32\] This interesting impact on the SWCNT photophysics could be attributed to the peptides folding differently on the SWCNT and changing the charge landscape through which the excitondiffuses or where it gets trapped, thus leading to enhanced \( E_{11}^* \) fluorescence for less hydrophobic sequences. Different folding of peptides is known from noncovalent SWCNT/peptide hybrids.\[61\] Furthermore, a comparison of the sequences 9–12 consisting of identical amino acids shows that not only the nature of the attached amino acids, but also their sequential arrangement is of high importance for the SWCNT* NIR fluorescence properties. These results demonstrate the possibilities of employing Fmoc-protected phenylalanine defects for the growth of peptide chains directly on the carbon nanotube sidewall and indicate that this method can not only be used for modulation of the SWCNTs’ fluorescence, but also to tailor their surface properties. This in turn could enable SWCNTs with enhanced cellular uptake/retention,\[52,53\] tailored molecular recognition motifs, and novel and more stable optical sensors operating in the NIR. Furthermore, the coupling of a second optically active molecule (fluorophore) via a peptide sequence to a SWCNT could serve as general design principle for molecular recognition and signal transduction. Similar to Förster resonance energy transfer (FRET), conformational changes upon binding to a target structure could affect the SWCNTs’ NIR fluorescence and enable novel fluorescent probes and labels.

**Conclusion**

In summary, we introduced two new sp\(^3\) quantum defects in SWCNTs, which serve as anchors for the attachment of biomolecules. The versatility of this new functionalization platform was demonstrated by conjugation of a GFP-binding nanobody as an example for a protein and the synthesis of peptides directly on the carbon nanotube surface. This new technique for covalent decoration of SWCNTs with biomolecules opens up great possibilities for applications in (bio)photonics, biosensing, and biomedicine.

**Acknowledgements**

We thank the VW foundation and the State of Lower Saxony (life@nano) for funding. We thank Andreas Janshoff and Claudia Steinem for their support. Open access funding enabled and organized by Projekt DEAL.
Conflict of interest

F. Opazo is shareholder of NanoTag Biotechnologies GmbH; all other authors declare no conflict of interest.

Keywords: bioconjugation · nanobodies · nanotubes · peptides · quantum defects
[54] F. A. Mann, J. Horlebein, N. F. Meyer, D. Meyer, F. Thomas, S. Kruss, Chem. Eur. J. 2018, 24, 12241 – 12245.

[55] G. R. Dieckmann, A. B. Dalton, P. A. Johnson, J. Razal, J. Chen, G. M. Giordano, E. Muñoz, I. H. Musselman, R. H. Baughman, R. K. Draper, J. Am. Chem. Soc. 2003, 125, 1770 – 1777.

[56] E. Polo, T. T. Nitka, E. Neubert, L. Erpenbeck, L. Vuković, S. Kruss, ACS Appl. Mater. Interfaces 2018, 10, 17693 – 17703.

[57] A. Bianco, K. Kostarelos, C. D. Partidos, M. Prato, Chem. Commun. 2005, 571 – 577.

[58] D. Pantarotto, C. D. Partidos, R. Graff, J. Hoebeke, J.-P. Briand, M. Prato, A. Bianco, J. Am. Chem. Soc. 2003, 125, 6160 – 6164.

[59] R. Haggenmueller, S. S. Rahatekar, J. A. Fagan, J. Chun, M. L. Becker, R. R. Naik, T. Krauss, L. Carlson, J. F. Kadla, P. C. Trulove, et al., Langmuir 2008, 24, 5070 – 5078.

[60] T. Shiraki, Y. Niidome, F. Toshimitsu, T. Shiraiishi, T. Shiga, B. Yu, T. Fujigaya, Chem. Commun. 2019, 55, 3662 – 3665.

[61] D. A. Heller, G. W. Pratt, J. Zhang, N. Nair, A. J. Hansborough, A. A. Boghosian, N. F. Reuel, P. W. Barone, M. S. Strano, Proc. Natl. Acad. Sci. USA 2011, 108, 8544 – 8549.

[62] M. Gravely, M. M. Safae, D. Roxbury, Nano Lett. 2019, 19, 6203 – 6212.

[63] D. Meyer, S. Telele, A. Zelená, A. J. Gillen, A. Antonucci, E. Neubert, R. Nüller, F. A. Mann, L. Erpenbeck, A. A. Boghosian, et al., Nanoscale 2020, 2, 751.

Manuscript received: March 14, 2020
Revised manuscript received: April 24, 2020
Accepted manuscript online: June 8, 2020
Version of record online: July 13, 2020