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Characterization and applications of exopolysaccharide produced by marine Bacillus altitudinis MSH2014 from Ras Mohamed, Sinai, Egypt

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1. Introduction

Exopolysaccharides (EPSs) are long chains of high molecular mass polymers produced by different microorganisms, including bacteria, fungi and blue green algae [1]. EPSs produced by bacteria exhibit significant structural diversity with novel biological properties are considered as valuable sources of natural polymers with multiple biotechnological applications. Microbial EPSs are prevalent in the extreme marine environment where they are essential for microbial existence. Most of the functions attributed to EPSs are of a protective nature and their accurate roles are dependent on the ecological functions in which the microorganisms live. They could support the microbial communities to suffer extremes of temperature, salinity and nutrient accessibility, construct a boundary between the bacterial cell and its included environment.

Bacteria in marine environments were forced with osmotic stress to produce EPSs with unique composition were studied for prospective application in various sectors [2] and for discovery of novel macromolecules [3]. Recently, evidences for antioxidant, immune-modulation, antitumor and antimicrobial properties of EPSs producing microorganisms have increased [4,5]. Serious side effects of clinically used antitumor drugs direct the attention towards investigation of novel agents with a higher potency and fewer effects [6]. Human pathogens, as well, have resistance formed from the abnormal use of commercial antimicrobial drugs. This resistance with the undesirable side effect of certain antibiotics motivated the scientists to look for new antimicrobial substituent from many sources [7].

The aim of this study is to deal with the isolation and characterisation of EPS producing bacteria from the marine sediments around the mangrove trees in Ras Mohamed area, Red Sea Coast, Sinai Peninsula, Egypt. The sample was serially diluted in 90 ml sterile water and plated on a medium containing (in g/l) [glucose (20); yeast extract (0.1); NH4NO3 (0.8); CaCO3 (1); K2HPO4 (0.6); KH2PO4 (0.5); MgSO4·7H2O (0.05), MnSO4·4H2O (0.1) and agar (15)] which dissolved in 750 ml seawater and 250 ml distilled water [8]. EPSs producing bacterial isolates were selected based on colonies phenotype (smooth and mucoid).

2. Materials and methods

2.1. Isolation of bacteria and production of EPSs

Marine sediment was collected from the rhizosphere of a mangrove trees (Avicennia marina) from Ras Mohamed, Sinai Peninsula, Egypt. The sample was serially diluted in 90 ml sterile water and plated on a medium containing (in g/l) [glucose (20); yeast extract (0.1); NH4NO3 (0.8); CaCO3 (1); K2HPO4 (0.6); KH2PO4 (0.5); MgSO4·7H2O (0.05), MnSO4·4H2O (0.1) and agar (15)] which dissolved in 750 ml seawater and 250 ml distilled water [8]. EPSs producing bacterial isolates were selected based on colonies phenotype (smooth and mucoid).
Pure strains were then inoculated in 250 ml conical flasks containing 50 ml of previous medium and incubated at 37 °C in a rotary shaker at 120 rpm for 48 h. After centrifugation at 5000 rpm for 15 min, supernatant was mixed with four volumes of chilled ethanol, while the collected pellets were washed twice with acetone and diethyl ether [5] and then drying at 50 °C to obtain constant weight. EPSs formation were estimated by the phenol-sulphuric acid method [9].

2.2. Identification of the strain

The strain with the highest EPS production was identified according to their morphology, physiology and biochemistry features such as nitrate reduction, catalase, oxidase, Voges-Proskauer test, acid production from glucose [10] and 16S rRNA sequence. The 16S rRNA gene was amplified and sequenced using universal primers described by Weisburg et al. [11]. The sequences were compared with the GenBank data base using BLAST [12].

2.3. Extraction and purification of EPS

The selected strain was grown in a fermentation medium containing (g/l) [peptone (4.0), yeast extract (2.0), and sucrose (20.0)] which dissolved in 750 ml seawater and 250 ml distilled water (pH 7–7.5) at 37 °C for three days in an incubator shaker at 120 rpm to produce EPS [13]. To precipitate proteins, 5% trichloroacetic acid was added to cell free culture supernatant and incubated at 4 °C for 24 h. After centrifugation at 5000 rpm for 15 min, supernatant pH was adjusted to 7 and dialyzed against distilled water using dialysis tube (MWCO 2000). After dialysis, abso- lute ethanol were added to supernatant and incubated at 4 °C for 24 h. The precipitated EPS was dissolved, dialysed against deionized water and filtered using 0.45 μm filter, then applied to a DEAE-cellulose column (1.5 cm × 70 cm). Elution was carried out using continuous gradient NaCl solution (0.2–3.0 M). Subsequently, a further purification step was carried out for the collected fractions using a Sephadex G-200 column (2 cm × 80 cm) which was eluted with 0.1 M NaCl at a flow rate of 0.5 ml/min. The fractions with EPS were collected in one fraction which dialyzed and lyophilized for further analysis.

2.4. Chemical analyses of EPS

Protein, uronic acid and monosaccharide sulphate were estimated for the produced EPS according to methods described by many authors [10,14–16] respectively. A high performance liquid chromatography (HPLC) was used to analyse the monosaccharide content using a Shimadzu Shim-Pack SCR-101N column (7.9 mm × 30 cm) and deionized water as mobile phase at flow rate 0.5 ml/min utilizing acrid hydrolysis technique [17].

2.5. Molecular weight estimation

The molecular weight of EPS was estimated by gel permeation chromatography (GPC) on a Sephadex G-200 column (80 cm × 2 cm). Standard dextrans (40,000; 500,000 and 2,000,000) Daltons were used for calibration and the molecular weight was determined by plotting against standard graph [18].

2.6. Fourier-transform infrared spectroscopy (FT-IR)

The dried EPS was mixed to KBr powder, ground and pressed into a 1 mm pellets for FT-IR measurements in the range of 400–4000 cm⁻¹ using Bruker scientific FT-500-IR spectrophotometer [19].

2.7. Periodate oxidation and Smith degradation

Periodate oxidation and Smith degradation were determined by dissolving desalted EPS (16 mg) in distilled water (6 ml) and then added to 0.1 M NaO4 (50 ml) in round bottom flasks which kept at 4 °C in dark [20]. Aliquots (0.1 ml) were taken every 24 h and diluted with distilled water (25 ml) which read in spectrophotometer at 223 nm.

Periodate consumption was calculated based on absorbance change [21] and formic acid products were monitored by the phenol-sulphuric acid method [9]. Ethylene glycol (2 ml) was added and the solutions were dialyzed against distilled water for 48 h. After that NaBH4 (100 mg) is added and the mixture was left for 24 h temperature and then ice cold acetic acid (4N) is added to stop the reaction. The solutions were again dialyzed as described above and lyophilized [22]. The resulting polyalcohol was hydrolyzed with HCOOH 90% for 5 h and the produced sugars and sugar alcohols were analyzed by HPLC.

2.8. Biological activity of EPS

2.8.1. Radical scavenging activity of EPS towards DPPH radical

The free radical scavenging activity of EPS was estimated according to method describing by Yang et al. [23] using 1,1-diphenyl-2-picryl-hydrazyl (DPPH). Five ml of DPPH in ethanol was added to 1 ml of purified EPS with concentrations (50–300 μg/ml). The mixture was incubated in dark for 30 min. and measured at 517 nm using spectrophotometer UV–Visible 2401PC (Shimadzu, Japan).

The DPPH absorbance goes down with a high free radical scavenging ability. The free radicals scavenger capacity was calculated based on this equation: Scavenging ability (%) = [(A517 of control – A517 of sample)/A517 of control] × 100. The EC50 value is considered as the effective concentration (μg) of EPS at which the DPPH absorbance was reduced by 50%.

2.8.2. Antitumor activity against Ehrlich Ascites Carcinoma Cells (EACC)

Tumor cells viability was measured with modified cytoxicity trypan blue exclusion technique as described by Yang et al. [23]. Tumor cells proliferation inhibition was calculated based on this equation: [(A–B/A) × 100], where A and B are the average number of viable tumor cells of the control and the samples, respectively.

2.8.3. Antitumor activity against lung cancer cell line A-549

Cytotoxicity of purified EPS was measured against A-549 cell line using the MTT cell viability assay [24]. Percentage of relative viability was calculated using the equation (Absorbance of treated cells/Absorbance of control cells × 100). Then the half maximal inhibitory concentration IC50 was calculated from the graph and BJ-1 normal human cell line was used as control.

2.8.4. Antimicrobial activity

The antimicrobial potential of the purified EPS was tested against a wide set of microorganisms including: (i) Gram positive bacteria (Bacillus subtilis NRRL B-941 and Staphylococcus aureus NRRL B-767), (ii) Gram negative bacteria (Pseudomonas aeruginosa NRRL B-23 and Escherichia coli NRRL B 210), (iii) yeast (Saccharomyces cerevisiae Y-2034 and Candida albicans NRRL Y-477) and (iv) fungi (Aspergillus niger NRRL-3 and Fusarium oxysporum NRRL 26406). The inhibition zones produced by different concentration of EPS (75, 100, 150, and 200 μg/disc) were determined. Rimactane as antibacterial (200 μg/disc) and flucosal as antifungal (200 μg disc) were used as control [25].
3. Results and discussion

3.1. Isolation, screening, and identification of the EPS producing bacteria

Many marine bacteria produce EPSs for their growth to adhere to solid surfaces and to overcome extreme conditions. In the present study, 29 bacterial strains were isolated from the rhizosphere sediment of mangrove trees in Ras Mohamed area, Red Sea; Sinai, Egypt. Two of them had the ability to produce EPSs. The highest production of EPS (10.33 g/l) was detected at strain No. 12. Liu et al. [26] reported in a study that 19 strains of bacteria produced EPSs in 2.28–9.02 g/l, where they selected a strain because of its highest EPS production and identified it as Bacillus licheniformis.

Colonies of strain No. 12 showed mucoid appearance on solid medium. The cells were Gram positive, rod shaped, motile and spore forming when observed under phase contrast microscope. In nitrate reduction, Voges-Proskauer tests and acid production from glucose, catalase and oxidase tests, the strain was positive, while negative for gas production from glucose. Hence, strain No. 12 was temporarily identified as Bacillus sp. particularly, the partially sequenced 16S rRNA genes showed 99% similarity with Bacillus altitudinis, so strain No. 12 was identified as Bacillus altitudinis MSH2014 with accession No. (KY550404) (Fig. 1).

3.2. Characterisation of EPS

3.2.1. Purification of EPS

The crude EPS produced from Bacillus altitudinis MSH2014 were purified using DEAE-cellulose anion exchange chromatography column (Fig. 2). Elution profile showed one major peak (fraction 100–120). Such fraction was collected, dialyzed and lyophilized to get EPS which applied for further analysis.

3.2.2. Molecular weight determination of EPS

The molecular weight of the EPS produced by Bacillus altitudinis MSH2014 determined by a gel filtration technique was $4.23 \times 10^5$ Da. EPS appeared as a single symmetrical narrow peak (Fig. 3), which represent the homogeneity of EPS. Previous reports of the molecular weight of the EPS produced by Bacillus licheniformis was $2.826 \times 10^4$ Da [26].

3.2.3. Chemical composition of EPS

The purified EPS produced a negative response by Bradford test indicating the absence of protein. Monosaccharide analysis by HPLC revealed that EPS composed of mannuronic acid and glucose with molar ratio 1:2.2. The chemical composition of EPS indicated the presence of uronic acid (14.26%) and sulphate (15.47%).
It was found that bacteria isolated from deep sea which produced EPS was typically acidic due to containing 10–40% uronic acid [27] and polyanionic due to uronic acids, ketal-linked pyruvate or inorganic residues such as PO₄ or SO₃ existence [28]. According to the literatures, the presence of SO₃ group enhanced the medical potential of EPS [29]. The sugar composition had an important role in the biological activities [30].

The FT-IR spectrum of EPS from *Bacillus altitudinis* MSH2014 showed a typical pattern for polysaccharide absorbance (Fig. 4). The broad band at 3435.6 cm⁻¹ is due to –OH groups of carbohydrate residues stretching vibration [31]. The 2360.0 cm⁻¹ band is due to C=H stretching vibration. The weak absorbance peak at 1630 cm⁻¹ was due to the stretching vibration of C=O that may be associated with the mannanuronic acid and internal hydrogen bonds [32]. Band at 1340 cm⁻¹ is due to the presence of S=O and/or C–O–S bonds [33] and the band at 863 cm⁻¹ indicates the α-configuration which is simultaneously present in EPS.

### 3.3. Periodate oxidation and Smith degradation

Periodate oxidation was done to both sulphated and desulfated EPS where in case of sulphated EPS did not produce formic acid and consumed periodate due to the presence of sulphate groups on the carbon atom 2 or 3 which impedes the process of periodate oxidation. But desulfated exopolysaccharide consumed periodate (0.942 mol) to produce formic acid (0.000801 mol) per one mole of anhydrosugar. HPLC analysis of EPS from *Bacillus altitudinis* MSH2014 shows erythritol, glycerol, and erytheric acid presence (Table 1). The presence of a small amount of glycerol and free erythritol in large amount partially demonstrate existence of (1 → 4) linkages between monosaccharides units and sulphate groups on the C2 and/or C3 in EPS. Erythritol was produced from C3, C4, C5 and C6 of the (1 → 4) glycosidic linkages of glucose after hydrolysis of the backbone, while erytheric acid was liberated from C3, C4, C5 and C6 of the (1 → 4) glycosidic linkage of mannanuronic acid. The presence of relatively small quantities of glycerol from the glucose units and erytheric acid from the mannanuronic unit in the hydrolysis of polysaccharides, gave the information that glucose may be found at the non-reducing end. These outcomes are in good agreement with those mentioned by other workers [34,35].

| Sugar alcohol | Erythritol | Glycerol | Erytheric acid |
|--------------|------------|----------|---------------|
| 3.3          | 0.4        | 0.9      |

Fig. 3. Gel filtration chromatography of EPS from *Bacillus altitudinis* MSH2014.

Fig. 4. FT-IR spectrum of the EPS from *Bacillus altitudinis* MSH2014.
3.4. Biological activity of EPS produced by Bacillus altitudinis MSH2014

3.4.1. Free radical scavenging activities of EPS

The free radical scavenging activity was estimated with DPPH [36]. The scavenging activity of EPS illustrates in Fig. 5 showed a concentration-dependent manner, with an EC_{50} value of 150 (μg/ml). It was reported that EPS was isolated from Bacillus coagulans RK-02 and purified by size exclusion chromatography which showed in vivo antioxidant activities [37]. Agili and Mohamed [38], also, mentioned that EPSs from Padina pavonica had antioxidant activities. The antioxidant properties of polysaccharides are mainly correlated to their confirmation, structure, molecular mass and their monosaccharides components [39].

3.4.2. Antitumor activity

3.4.2.1. Antitumor activity against Ehrlich Ascites Carcinoma Cells (EACC). As shown in Fig. 6, the EPS exerted inhibitory activity on EACC in concentration dependent manner. The viability of tumor cells after incubation with purified EPS drastically decreased from 47 to 20% and then gradually decreased further to 10% by increasing the concentration of EPS from 100 to 500 till 1500 μg/ml respectively, which is in agreement with the results obtained by Ahmed and Ahmed [40].

3.4.2.2. In vitro antitumor activity against lung cancer cell. Lung cancer cell line A-549 was treated with the purified EPS at different concentrations (12.5, 25, 50 and 100 μg/ml) and showed in Fig. 7. The data shows a gradual decrease in tumor cells viability with high concentrations of EPS. The highest cells viability (71%) was found at 12.5 μg/ml which decreases to 63%, 51% and 39% at 25 μg/ml, 50 μg/ml, 100 μg/ml of EPS, respectively, and IC_{50} was 51.94 μg/ml. While in case of normal cell line the viability of cells reached 95%, 90%, 88% and 77% when treated with the different concentrations of EPS 12.5, 25, 50 and 100 μg/ml respectively.

The differences in the antitumor activities of EPS may be due to their different physicochemical properties such as chain shape, molecular weight and monosaccharide composition [41]. EPS molecular weight may affect its bioactivities, as the high molecular weight showed more antitumor activity [42].

3.4.3. Antimicrobial activities

The antimicrobial activities (Table 2) demonstrate that the purified EPS possessed antimicrobial activities against all of the tested microorganism and by increasing the concentration of EPS the inhibition zone increased which ranged from 6.3 to 24.9 mm. Orsod et al. [43] isolated two EPS producers marine bacteria and screened their antimicrobial activities against both Gram positive bacteria (Lysinibacillus and Paenibacillus sp.) and Gram negative bacteria (Pseudomonas sp., Escherichia coli) had shown presence of inhibition zones.

The mechanism of the antimicrobial activity of the purified EPSs were studied by various investigators and they mentioned that the

| Concentrations of EPS (μg/disk) | Gram+ve bacteria | Gram–ve bacteria | Yeast | Fungi |
|--------------------------------|-----------------|-----------------|-------|------|
|                                | B. subtilis     | St. aureus     | E. coli | P. aeruginosa | S. cerevisiae | C. albicans | A. niger | F. oxysporum |
| 75                              | 11.2            | 12.2            | 12.9  | 7.7  | 10.2 | 7.7 | 15.2 | 6.3 |
| 100                             | 13.1            | 15.1            | 17.7  | 10.6 | 12.2 | 9.5 | 16.7 | 7.1 |
| 150                             | 15.7            | 17.3            | 19.8  | 13.4 | 14.7 | 13.7 | 18.7 | 8.3 |
| 200                             | 17.8            | 18.8            | 24.9  | 15.6 | 17.6 | 17.3 | 20.0 | 10.5 |
| Rimactane                       | 16.7            | 16.8            | 21.9  | 14.3 | 0.0  | 0.0  | 0.0  | 0.0  |
| Flucoral                        | 0.0             | 0.0             | 0.0   | 0.0  | 25.9 | 24.1 | 24.5 | 26.4 |

Table 2 Inhibition zones of microbial growth by different concentrations of the EPS from Bacillus altitudinis MSH2014.
antimicrobial activity of anionic polysaccharides such as sulphated polysaccharide occurs by several mechanisms through its chelation activities and the deprivation of metal, trace elements or essential nutrients which limit the growth of microorganisms [44].

4. Conclusion

Although EPSs are one of the abundant bioactive components, their study still remains an interesting scope for their high molecular diversity, complexity and high capability for drugs design and preparation. In addition, microorganisms producing EPSs are promising sources that meet actual industrial demand, especially, marine microorganisms which offer great opportunities for new bioactive compound discovery.

Conflict of interest statement

The authors declare that there is no conflict of interest with any financial organization or corporation or individual that can inappropriately influence this work.

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