Subcellular Metabolite and Lipid Analysis of *Xenopus laevis* Eggs by LAESI Mass Spectrometry

Bindesh Shrestha¹, Prabhakar Sripadi¹, Brent R. Reschke³, Holly D. Henderson³, Matthew J. Powell³, Sally A. Moody², Akos Vertes¹*

¹. Department of Chemistry, W. M. Keck Institute for Proteomics Technology and Applications, The George Washington University, Washington, D.C., United States of America, 2. Department of Anatomy and Regenerative Biology, The George Washington University, Washington, D.C., United States of America, 3. Protea Biosciences, Morgantown, West Virginia, United States of America

*vtes@gwu.edu

These authors contributed equally to this work.

Current address: National Centre for Mass Spectrometry, CSIR-Indian Institute of Chemical Technology (IITC), Tarnaka, Hyderabad, India

Abstract

*Xenopus laevis* eggs are used as a biological model system for studying fertilization and early embryonic development in vertebrates. Most methods used for their molecular analysis require elaborate sample preparation including separate protocols for the water soluble and lipid components. In this study, laser ablation electrospray ionization (LAESI), an ambient ionization technique, was used for direct mass spectrometric analysis of *X. laevis* eggs and early stage embryos up to five cleavage cycles. Single unfertilized and fertilized eggs, their animal and vegetal poles, and embryos through the 32-cell stage were analyzed. Fifty two small metabolite ions, including glutathione, GABA and amino acids, as well as numerous lipids including 14 fatty acids, 13 lysophosphatidylcholines, 36 phosphatidylcholines and 29 triacylglycerols were putatively identified. Additionally, some proteins, for example thymosin β4 (Xen), were also detected. On the subcellular level, the lipid profiles were found to differ between the animal and vegetal poles of the eggs. Radial profiling revealed profound compositional differences between the jelly coat vitelline/plasma membrane and egg cytoplasm. Changes in the metabolic profile of the egg following fertilization, e.g., the decline of polyamine content with the development of the embryo were observed using LAESI-MS. This approach enables the exploration of metabolic and lipid changes during the early stages of embryogenesis.
Introduction

The early stages of embryonic development following fertilization in animals are characterized by synchronous cell divisions, the onset of transcription of genes that will pattern the embryo, and local signaling events that transition a ball of equipotent cells into regions that express different tissue fates [1, 2]. These events are common across different species of animals, but can occur for different lengths of times and at different stages of morphogenesis.

The African clawed frog, *Xenopus laevis*, has been used extensively to study the cellular and molecular events of early embryogenesis. This species is used because they are easily bred in the laboratory, are disease resistant, have a reproductive response that is independent of season, and produce a large number of eggs per spawning. In particular, the large size of the oocytes, ova (unfertilized eggs) and embryos make them easy to manipulate and analyze by standard biochemical and molecular biology approaches [3, 4, 5, 6, 7].

*X. laevis* oocytes and eggs have been an important source of material for elucidating the molecular regulation of the cell cycle. For example, traditional protein chemistry approaches led to the identification of maturation promoting factor and the cyclin proteins [8, 9]. More recently, the oocyte has been used to identify the proteins involved in nuclear transport and DNA replication [6]. To reveal the biochemical makeup, elucidate low abundance proteins, and study the kinetics of developmental and cellular events, *X. laevis* oocytes have been analyzed by high performance liquid chromatography (HPLC) [10, 11][12], time-of-flight (TOF) secondary ion mass spectrometry (SIMS) [13], gas chromatography (GC) mass spectrometry (MS) [14], 2D gel electrophoresis followed by electrospray ionization (ESI) MS [15], and liquid chromatography-tandem MS (LC-MS/MS) [2]. To study the types and amounts of lipids extracted from *Xenopus* eggs, HPLC methods were developed for the detection of multiple species [10, 12]. Similarly, with an HPLC method for characterizing the carbohydrates in the jelly coat of *Xenopus* eggs, specific oligosaccharides were identified and assigned to the three specific layers of the coat [11, 16]. In another study, LC-MS/MS was used to monitor the amounts of alanine in the early embryo, resulting in a new hypothesis for the increase in cell cycle time at the mid-blastula transition [2]. Using a GC-MS method for analyzing metabolites from volume-limited samples, e.g., *X. laevis* eggs, good reproducibility was shown for volumes as small as 10 nL. However, the method required silylation of some metabolites to increase their volatility for analysis using GC [14]. On a subcellular scale, 3D molecular images of freeze dried *Xenopus* oocytes were obtained under optimized sample preparation conditions using TOF-SIMS [13]. While all of these methods for *X. laevis* oocytes and embryo analysis have been shown to capture important aspects of its biochemistry, most of them require elaborate sample preparation prior to analysis that might alter subcellular localization of important developmental molecules.

An advantage of direct analysis methods is that they can report on the unperturbed composition of biological specimens, ideally in their living, natural state. Recently, the analysis of single oocytes and embryos of mice and bovine...
species by desorption electrospray ionization (DESI) mass spectrometry provided new insight into lipid metabolism changes during early embryonic development [17, 18].

In another direct analysis technique, laser ablation electrospray ionization (LAESI) MS, the energy from a focused mid-infrared laser pulse is absorbed by the water molecules in a cell or tissue leading to the ablation sampling of adjacent cellular content [19]. Each consecutive laser pulse ablates a successive layer of the sample enabling analysis at a defined subcellular depth or of an entire cell [20, 21]. Combination of LAESI with microdissection of plant cells reveals subcellular metabolite gradients between the cytoplasm and the nucleus [22]. Diverse applications of LAESI-MS include the analysis of virally infected immune cells as well as comparative analysis of plant cell phenotypes [23, 24]. Because the developing egg is a spatially and temporally dynamic system that is sensitive to environmental and thus preparatory changes, we used the direct LAESI-MS analysis of individual X. laevis eggs and embryos to explore subcellular asymmetries and potential changes during significant developmental transitions. Using LAESI-MS to probe these specimens required minimal sample preparation, was performed at atmospheric pressure, and each ablation event occurred instantaneously (in less than a second) on the developmental time scale. The analyses were performed on living eggs and embryos that were minimally perturbed from their natural state.

Some of the metabolic profiles of the unfertilized egg obtained from the LAESI-MS analyses were compared to results reported previously in the literature and showed good agreement, thus validating the approach. In addition we identified subtle differences between the unfertilized egg, in which the first meiotic division is completed, and the fertilized egg and early cleavage stage embryos. Importantly, we found differences between the animal and vegetal poles of the egg, which are predisposed to different developmental fates. Since these early developmental events are shared across animals, a direct analysis of the cellular constituents should provide fundamental information about the common biochemical processes that regulate them. This simple approach using a single, live embryo in its natural state should be ideal for analyses of other embryos that are difficult to collect and are small in size.

Materials and Methods

Ova and Embryos

X. laevis eggs and sperm were obtained from mature animals kept in 30 gallon aquaria at 21°C in natural daylight and fed frog brittle 2 times per week. Oocyte maturation was induced by injecting frogs with 1000 U human chorionic gonadotropin (Sigma) 12 hours prior to removal of eggs as described elsewhere [25]. Mature eggs were gently squeezed from female frogs into Petri dishes; some were collected prior to fertilization and others were fertilized in vitro and collected at various stages after sperm entry. Sperm were obtained from minced testes of
sacrificed males that were anesthetized by submersion in an ice bath with tricaine methanesulfonate. Eggs were fertilized in vitro by adding the released sperm to the eggs and adding $0.1 \times$ MBS ($1 \times = 88$ mM NaCl, $1$ mM KCl, $1$ mM MgSO$_4$, $0.7$ mM CaCl$_2$, $5$ mM HEPES, pH 7.8 and $2.5$ mM NaHCO$_3$) [26]. The jelly coat was removed from some samples by washing the eggs with 2% cysteine solution (pH 8.0, adjusted by 10 M NaOH) for 4 minutes followed by washing in $0.1 \times$ Steinberg’s solution ($1 \times = 60$ mM NaCl, $0.67$ mM KCl, $0.83$ mM MgSO$_4$, $0.34$ mM Ca(NO$_3$)$_2$, $4$ mM Tris-HCl, $0.66$ mM Tris Base and pH 7.4). For some experiments the vitelline membrane was manually removed from each egg using sharpened forceps. Unfertilized eggs of X. laevis were also purchased from Xenopus Express, Inc. (Brooksville, FL) for some confirmatory studies. All animals used in these experiments followed the U.S. Public Health Service Policy of Humane Care and Use of Laboratory Animals and were approved by the George Washington University (GWU Animal Study Protocol: #A-3205-01; A237).

Chemicals
All chemicals for egg preparation were reagent grade and obtained from Sigma. For the electrospray solution, highly purified spectral-grade methanol and water solvents were obtained from Acros Organics (Geel, Belgium), and glacial acetic acid was purchased from Fluka (Munich, Germany). All of the chemicals were used without further purification.

LAESI-MS
The LAESI ion source used was similar to those described earlier [19]. Briefly, mid-infrared laser pulses at 2940 nm wavelength and 20 Hz repetition rate were produced by a Nd:YAG laser driven optical parametric oscillator system (Opolette 100, Opotek, Carlsbad, CA). Laser pulses with an average 320 $\mu$L/pulse energy were focused to about 200–300 $\mu$m diameter spot with either a 150 mm focal length CaF$_2$ lens or a 75 mm ZnSe lens. The eggs/embryos were placed on a pre-cleaned microscope glass slide held 15 mm below the electrospray axis that was in line with the mass spectrometer orifice. In the home-built electrospray system, a syringe pump supplied 50% methanol solution containing 0.1% acetic acid through a tapered tip metal emitter (length 5 cm, tube OD 320 $\mu$m and tip ID 50 $\mu$m) at a flow rate of 300–400 nL/min. The electrospray emitter was held at 2800–3300 V. Positive ions were collected by either an orthogonal acceleration TOF mass spectrometer (Q-TOF Premier, Waters Co., Milford, MA) with a mass range of $m/z$ 50–2000 and a typical resolution of 8,000 (FWHM) or with a high performance TOF mass spectrometer (Synapt G2S, Waters Co., Milford, MA) with a mass range of $m/z$ 50–2000 and a typical resolution of 30,000 (FWHM) [27].
Data Analysis
The electrospray background ion signal was subtracted from the collected LAESI mass spectra in the MassLynx 4.1 software (Waters Co., Milford, MA). Putative peak assignments for metabolites and lipids were based on accurate mass measurements, isotope distribution patterns, database searches, data mining of the related literature, and in some cases tandem MS analysis. Multiple databases were used for the metabolite and lipid searches, including METLIN [28], MetaCyc [29], Lipid Maps [30], HMDB [31] and KEGG [32, 33]. They were last accessed on August 12, 2014. The NIST Isotope Calculator program (ISOFORM, Version 1.02) was used to calculate monoisotopic masses.

Results and Discussions

Ovum Analysis
A single *X. laevis* egg is ∼1.4 mm in diameter and has an approximate volume of 400 nL. A typical LAESI mass spectrum from a single unfertilized *X. laevis* egg without the removal of its jelly coat is shown in Fig. 1. Most of the ions below m/z 450 correspond to small metabolites, whereas many ions between m/z 450 and 900 are assigned to lipids. For a typical unfertilized *X. laevis* ovum, putative assignments of the ions generally found in the m/z ≤550 region reveals 52 small metabolites (see Table 1). Many of these metabolites, e.g., amino acids, organic acids, and redox buffering agents, fulfill an essential role in cell development and serve as building blocks for cellular biosynthesis. Using liquid extraction and LC-MS/MS, a previous study followed 48 metabolites in *X. laevis* embryos [2]. In Table 1, sixteen metabolites found by both the LC-MS/MS and LAESI-MS methods are marked by asterisks. The advantage of LAESI-MS over conventional techniques is that it rapidly (within seconds) identifies metabolites in a single egg with minimum perturbation before analysis. It also provides complementary data for improved coverage of the small metabolites.

In the mass spectra we noticed the presence of multiply charged peaks at m/z 835, 1,002, and 1,251 with charge states of 6+, 5+ (see Fig. 1) and 4+, respectively. Deconvolution of these peaks revealed that they corresponded to thymosin β4 (Xen) (see the inset in Fig. 1). This 44-residue peptide in *Xenopus* has been highly conserved throughout evolution as it only differs from the human thymosin β4 in three amino acid residues [34]. A typical defolliculated oocyte contains 0.5 and 10 picomoles of thymosin β4 (Xen) in stage I and stage VI of the oogenesis, respectively [34]. Our results show that unfertilized eggs, the next step after stage VI oocytes, continue to contain thymosin β4 (Xen) in significant quantities.

Radial Profiling of Ovum
The *X. laevis* egg is surrounded by a jelly coat that must be penetrated by the sperm during fertilization. The LAESI mass spectra of the jelly coat were dominated by sodiated disaccharide, trisaccharide, and tetrasaccharide molecules.
at \( m/z \) 365.108, 527.192 and 689.286, respectively. The *Xenopus* egg jelly is known to contain a 21,073 Da protein, allurin, that serves as a chemoattractant for the sperm [35]. The LAESI mass spectra of the jelly coat did not show the ions for allurin, but exhibited strong signal for multiply charged ions corresponding to a protein with a molecular weight of 11,728 ± 1 Da (see Fig. 2) [36].

Radial profiling of an unfertilized *X. laevis* egg with a jelly coat was conducted by ablating the same area with multiple laser pulses, producing a mass spectrum for each shot (see Fig. 3). The first approximately six laser shots produced mass spectra similar to that of the jelly coat (see the top panel in Fig. 3). This observation was consistent with the ~30 \( \mu m \) ablation depth per laser shot and the ~200 \( \mu m \) thickness of typical jelly coats. Once the jelly coat-related signal diminished, multiply charged thymosin \( \beta 4 \) (Xen) peptide ions were detected (see the middle panel in Fig. 3). From its radial localization, it seems that this peptide is associated with the vitelline and/or plasma membranes. After approximately two additional laser shots, metabolites and lipid species characteristic to the egg cytoplasm composition were detected in the mass spectra (see bottom panel in Fig. 3).

**Lipid Composition of Unfertilized Ovum**

The cytoplasm of the unfertilized egg contains a high percentage of yolk platelets composed of proteins and lipids for nourishment of the embryonic cells [37]. The major peaks observed between \( m/z \) 400 and 900 in the LAESI mass spectrum are attributed to lipids. Earlier studies on phospholipid extracts from *X. laevis* eggs showed that the major lipid components were phosphatidylcholines (PC) at ~20 nmol/egg, followed by phosphatidylethanolamines (PE) at ~10 nmol/egg, phosphatidylinositol (PI) at ~4 nmol/egg, lysophosphatidylcholines (LPC) at...
Table 1. Putative identification of metabolite ions from LAESI-MS of an unfertilized *X. laevis* egg.

| Metabolite                          | Chemical formula | Monoisotopic mass | Measured mass | Δm (mDa) |
|-------------------------------------|------------------|-------------------|---------------|----------|
| carbamate                           | CH₂N₂O₂          | 82.998 (Na⁺)      | 82.997        | −1       |
| putrescine                          | C₆H₁₂N₂          | 89.108 (H⁺)       | 89.106        | −2       |
| alanine*                            | C₃H₇NO₂          | 90.056 (H⁺)       | 90.054        | −2       |
| choline*                            | C₂H₁₅NO           | 104.106 (H⁺)      | 104.106       | −1       |
| propionic acid                      | C₃H₆O₂            | 113.012 (K⁺)      | 113.012       | 11       |
| creatinine                          | C₄N₃H₇O          | 114.064 (H⁺)      | 114.064       | −3       |
| propanediol                         | C₃H₆O₂            | 115.016 (H⁺)      | 115.022       | 6        |
| proline*                            | C₂H₃NO₂          | 116.071 (H⁺)      | 116.071       | 0        |
| niacinamide                         | C₄H₇N₂O          | 138.053 (Na⁺)     | 138.053       | 0        |
| hydroxyethylphosphonate             | C₂H₅O₄P          | 127.016 (H⁺)      | 127.023       | 7        |
| butyric acid                        | C₄H₈O₂            | 127.016 (K⁺)      | 127.023       | 7        |
| octenol                             | C₆H₁₆O            | 129.128 (H⁺)      | 129.135       | 7        |
| creatine                            | C₅H₁₀N₂O₂         | 132.077 (H⁺)      | 132.075       | −2       |
| imidazole acetaldehyde (or)         | C₃H₇NO₂O          | 133.038 (Na⁺)     | 133.033       | −5       |
| N-acetylimidazole                   | C₆H₅N₂O          | 133.038 (Na⁺)     | 133.033       | −5       |
| malic acid*                         | C₄H₆O₅            | 135.029 (H⁺)      | 135.030       | 1        |
| adenine                             | C₅H₇N₂O          | 136.062 (H⁺)      | 136.053       | −9       |
| threonic acid or erythronic acid    | C₄H₈O₅            | 137.045 (H⁺)      | 137.044       | −1       |
| hexenal                             | C₆H₁₀O            | 137.037 (K⁺)      | 137.044       | 7        |
| valeric acid or pentanoic acid      | C₅H₁₀O₂           | 141.032 (K⁺)      | 141.038       | 6        |
| gamma aminobutyric acid amide       | C₅H₁₃N₂O₂         | 141.043 (K⁺)      | 141.038       | −5       |
| gamma-hydroxybutyric acid           | C₅H₁₅O₃          | 143.011 (K⁺)      | 143.018       | 7        |
| spermidine                          | C₆H₁₈N₃           | 146.166 (H⁺)      | 146.164       | −2       |
| guaiacol                            | C₆H₁₀O₂           | 147.042 (Na⁺)     | 147.048       | 6        |
| glutamate*                          | C₆H₁₀N₄O          | 148.061 (H⁺)      | 148.058       | −3       |
| methionine*                         | C₅H₁₁N₂O₂S        | 150.059 (H⁺)      | 150.047       | 12       |
| guanine* or hydroxyadenine          | C₅H₁₄N₃O          | 152.057 (H⁺)      | 152.054       | −3       |
|                                  |                  | 174.039 (Na⁺)     | 174.038       | −1       |
|                                  |                  | 190.013 (K⁺)      | 190.012       | −1       |
| dihydroorotic acid                  | C₅H₁₅N₂O₄         | 159.041 (H⁺)      | 159.040       | −0       |
| homospermidine                      | C₆H₁₃N₃           | 160.181 (H⁺)      | 160.180       | −1       |
| urocanic acid                       | C₆H₁₀N₂O₂         | 161.033 (Na⁺)     | 161.029       | −4       |
| carnitine*                          | C₇H₁₅NO₃          | 162.113           | 162.102       | 11       |
| aspartic acid*                      | C₄H₇NO₄           | 172.001 (K⁺)      | 172.008       | 7        |
| glycerol 3-phosphate*               | C₃H₈O₅P           | 173.022 (H⁺)      | 173.028       | 6        |
| arginine                            | C₆H₁₂N₂O₂         | 175.120 (H⁺)      | 175.117       | −3       |
| cys gly                             | C₅H₁₀N₂O₂S₁       | 179.049 (H⁺)      | 179.046       | −3       |
| phosphocholine                      | C₅H₁₄N₂O₄P       | 184.074 (H⁺)      | 184.070       | −4       |
| histidine*                          | C₅H₁₃N₂O₂         | 194.033 (H⁺)      | 194.033       | 0        |
| phosphocreatine                     | C₅H₁₂N₂O₅P       | 212.044 (H⁺)      | 212.042       | −2       |
| N-formylmethionine                  | C₅H₁₁NO₂S         | 216.010 (H⁺)      | 216.010       | 0        |
| citrate*                            | C₆H₁₂O₇           | 215.0168 (Na⁺)    | 215.010       | −7       |
| Metabolite                          | Chemical formula | Monoisotopic mass | Measured mass | Δm (mDa) |
|------------------------------------|------------------|-------------------|---------------|----------|
| dopaquinone                        | C₉H₉NO₄          | 234.017 (K⁺)      | 234.022       | 5        |
| homocarnosine or balenine          | C₁₀H₁₆N₂O₃       | 241.130 (H⁺)      | 241.131       | 1        |
| pantothenate*                      | C₃H₁₇NO₅         | 242.100 (Na⁺)     | 242.088       | -12      |
|                                   |                  | 258.074 (K⁺)      | 258.070       | -4       |
| methylthio propylmalic acid        | C₈H₁₄O₅S         | 245.046 (Na⁺)     | 245.050       | 4        |
|                                  |                  | 261.020 (K⁺)      | 261.027       | 7        |
| acetyldihydrolipoamide             | C₁₀H₁₂NO₂S₂      | 250.094 (H⁺)      | 250.091       | -3       |
| glutathione*                       | C₁₀H₁₇N₃O₆S     | 308.092 (H⁺)      | 308.090       | -2       |
|                                   |                  | 330.074 (Na⁺)     | 330.068       | -6       |
|                                   |                  | 346.048 (K⁺)      | 346.047       | -1       |
| hydroxydesipramine                 | C₁₈H₂₂N₂O       | 321.137 (K⁺)      | 321.127       | -10      |
| disaccharide or trehalose*         | C₁₂H₂₂O₁₁       | 365.106 (Na⁺)     | 365.107       | 1        |
|                                   |                  | 381.080 (K⁺)      | 381.080       | 0        |
| cholesterol                        | C₂₇H₄₆O          | 369.352 (H⁺, -H₂O)| 369.354       | 2        |
| adenosine monophosphate (AMP)      | C₁₀H₁₆N₂O₇P      | 386.027 (K⁺)      | 386.017       | -10      |
| adenosine diphosphate (ADP)*       | C₁₀H₁₇N₂O₁₀P₂    | 428.037 (H⁺)      | 428.032       | -5       |
|                                   |                  | 450.019 (Na⁺)     | 450.006       | -13      |
|                                   |                  | 465.993 (K⁺)      | 465.981       | -12      |
| adenosine triphosphate (ATP)*      | C₁₀H₁₆N₅O₁₃P₃   | 508.004 (H⁺)      | 508.002       | -2       |
|                                   |                  | 529.986 (Na⁺)     | 529.975       | -11      |

*Metabolites also detected in Ref. [2].

doi:10.1371/journal.pone.0115173.t001

Fig. 2. Positive ion LAESI mass spectrum of *X. laevis* egg jelly coat primarily shows sodiated oligosaccharide ions with a set of multiply charged ions at higher m/z corresponding to a species with molecular weight of 11728 ± 1 Da.

doi:10.1371/journal.pone.0115173.g002
2 nmol/egg, and phosphatidylserines (PS) at 1 nmol/egg [12]. In addition to phospholipids, triacylglycerols (TAG) were the major neutral lipid components found in the Xenopus eggs along with fatty acids (FA), cholesterol and diacylglycerols (DAG) [38].

The LAESI mass spectra gave additional insight into the nature of the lipids present in the ovum. A wide variety of lipid classes were identified, including 14 FAs, 13 LPCs, 36 PCs and 29 TAGs. Most of the abundant lipid peaks in these spectra corresponded to PCs, where one of the acyl chains was palmitic acid (16:0), whereas the other acyl chain length and saturation varied between (14:0) and (22:6) (see Table 2). The acyl chain length assignments were based on the tandem mass spectra of the lithiated lipids, [M+Li]+, generated by reactive LAESI. In this technique Li+ ions from the electrospray solution react with the lipid molecules from the ablated sample [39]. For example, tandem MS of a protonated lipid ion at m/z 760.582 produced a single fragment ion corresponding to its phosphocholine head group at m/z 184.073 resulting in the overall assignment of PC(34:1) showing only the combined length and number of double bonds for the acyl chains (see Fig. 4a). Further information on the individual acyl chains was obtained by producing and fragmenting the corresponding lithiated species.

~2 nmol/egg, and phosphatidylserines (PS) at ~1 nmol/egg [12]. In addition to phospholipids, triacylglycerols (TAG) were the major neutral lipid components found in the Xenopus eggs along with fatty acids (FA), cholesterol and diacylglycerols (DAG) [38].

The LAESI mass spectra gave additional insight into the nature of the lipids present in the ovum. A wide variety of lipid classes were identified, including 14 FAs, 13 LPCs, 36 PCs and 29 TAGs. Most of the abundant lipid peaks in these spectra corresponded to PCs, where one of the acyl chains was palmitic acid (16:0), whereas the other acyl chain length and saturation varied between (14:0) and (22:6) (see Table 2). The acyl chain length assignments were based on the tandem mass spectra of the lithiated lipids, [M+Li]+, generated by reactive LAESI. In this technique Li+ ions from the electrospray solution react with the lipid molecules from the ablated sample [39]. For example, tandem MS of a protonated lipid ion at m/z 760.582 produced a single fragment ion corresponding to its phosphocholine head group at m/z 184.073 resulting in the overall assignment of PC(34:1) showing only the combined length and number of double bonds for the acyl chains (see Fig. 4a). Further information on the individual acyl chains was obtained by producing and fragmenting the corresponding lithiated species.

~2 nmol/egg, and phosphatidylserines (PS) at ~1 nmol/egg [12]. In addition to phospholipids, triacylglycerols (TAG) were the major neutral lipid components found in the Xenopus eggs along with fatty acids (FA), cholesterol and diacylglycerols (DAG) [38].

The LAESI mass spectra gave additional insight into the nature of the lipids present in the ovum. A wide variety of lipid classes were identified, including 14 FAs, 13 LPCs, 36 PCs and 29 TAGs. Most of the abundant lipid peaks in these spectra corresponded to PCs, where one of the acyl chains was palmitic acid (16:0), whereas the other acyl chain length and saturation varied between (14:0) and (22:6) (see Table 2). The acyl chain length assignments were based on the tandem mass spectra of the lithiated lipids, [M+Li]+, generated by reactive LAESI. In this technique Li+ ions from the electrospray solution react with the lipid molecules from the ablated sample [39]. For example, tandem MS of a protonated lipid ion at m/z 760.582 produced a single fragment ion corresponding to its phosphocholine head group at m/z 184.073 resulting in the overall assignment of PC(34:1) showing only the combined length and number of double bonds for the acyl chains (see Fig. 4a). Further information on the individual acyl chains was obtained by producing and fragmenting the corresponding lithiated species.
| Lipid                                | Chemical formula | Monoisotopic mass | Measured mass | Δm (mDa) |
|--------------------------------------|------------------|-------------------|---------------|----------|
| Propionic acid (3:0)                 | C_3H_6O_2        | 113.001 (K⁺)      | 112.992       | -9       |
| Butenoic acid or Crotonic acid (4:1)| C_4H_6O_2        | 125.001 (K⁺)      | 124.998       | -3       |
| Butyric acid (4:0)                  | C_4H_8O_2        | 127.016 (K⁺)      | 127.010       | -6       |
| Pentanoic acid                      | C_5H_10O_2       | 141.032 (K⁺)      | 141.039       | 7        |
| Caprylic acid (8:0)                 | C_8H_16O_2       | 183.079 (K⁺)      | 183.072       | -7       |
| FA(14:0)                            | C_{14}H_{28}O_2  | 267.173 (K⁺)      | 267.167       | -6       |
| FA(15:0)                            | C_{15}H_{30}O_2  | 281.188 (K⁺)      | 281.184       | -4       |
| FA(16:1)                            | C_{16}H_{32}O_2  | 277.214(Na⁺)      | 277.200       | -14      |
|                                    |                  | 293.188 (K⁺)      | 293.179       | -9       |
| FA(16:0)                            | C_{16}H_{32}O_2  | 295.230 (Na⁺)     | 295.217       | -13      |
|                                    |                  | 295.195 (K⁺)      | 295.195       | -9       |
| FA(18:5)                            | C_{18}H_{36}O_2  | 297.183 (Na⁺)     | 297.179       | 11       |
| FA(18:4)                            | C_{18}H_{34}O_2  | 315.173 (Na⁺)     | 315.167       | -5       |
|                                    |                  | 319.204 (K⁺)      | 319.195       | -9       |
| FA(18:1)                            | C_{18}H_{34}O_2  | 321.220 (K⁺)      | 321.214       | -6       |
| FA(18:0)                            | C_{18}H_{36}O_2  | 323.235 (K⁺)      | 323.226       | -9       |
|                                    |                  | 327.230 (Na⁺)     | 327.222       | -8       |
|                                    |                  | 331.239 (K⁺)      | 331.226       | -9       |
| FA(20:5)                            | C_{20}H_{44}O_2  | 341.188 (Na⁺)     | 341.179       | -10      |
|                                    |                  | 343.204 (K⁺)      | 343.195       | -4       |
| FA(20:4)                            | C_{20}H_{44}O_2  | 351.230 (Na⁺)     | 351.217       | -9       |
|                                    |                  | 351.226 (K⁺)      | 351.214       | -6       |
| FA(22:6)                            | C_{22}H_{46}O_2  | 369.220 (K⁺)      | 369.210       | -10      |
|                                    |                  | 375.230 (Na⁺)     | 375.217       | 9        |
|                                    |                  | 367.204 (K⁺)      | 367.201       | -3       |
| FA(22:5)                            | C_{22}H_{46}O_2  | 369.220 (K⁺)      | 369.210       | -10      |
| LPC(14:0)                           | C_{22}H_{44}NO_7P| 468.309 (H⁺)      | 468.313       | 4        |
|                                    |                  | 482.325 (H⁺)      | 482.333       | 8        |
|                                    |                  | 494.325 (H⁺)      | 494.329       | 4        |
|                                    |                  | 518.322 (Na⁺)     | 518.319       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.319       | 7        |
| LPC(16:1)                           | C_{24}H_{46}NO_6P| 476.314 (H⁺, -H₂O)| 476.312      | -2       |
|                                    |                  | 496.340 (H⁺)      | 496.342       | 2        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
| LPC(16:0)                           | C_{24}H_{50}NO_7P| 476.314 (H⁺, -H₂O)| 476.312      | -2       |
|                                    |                  | 496.340 (H⁺)      | 496.342       | 2        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
|                                    |                  | 518.322 (Na⁺)     | 518.329       | 7        |
Table 2. Cont.

| Lipid          | Chemical formula | Monoisotopic mass | Measured mass | Δm (mDa) |
|----------------|------------------|-------------------|--------------|----------|
| LPC(18:1)      | C_{26}H_{52}NO_7P | 273.670 (H^+K)^2+ | 273.663      | -7       |
|                |                   | 280.660 (H^+K)^2+ | 280.656      | -4       |
| LPC(18:0)      | C_{26}H_{54}NO_7P | 504.345 (H^+-H_2O)^2+ | 504.347   | 2        |
| LPC(20:5)      | C_{26}H_{52}NO_7P | 506.361 (H^+-H_2O) | 506.360     | -1       |
|                |                   | 526.330 (H^+)     | 526.324     | -6       |
| LPC(20:4)      | C_{26}H_{52}NO_7P | 542.325 (H^+)     | 542.327     | 2        |
|                |                   | 524.314 (H^+-H_2O) | 524.320    | -6       |
| LPC(22:6)      | C_{26}H_{52}NO_7P | 568.340 (H^+)     | 568.348     | 8        |
| PA(O-37:1)     | C_{40}H_{79}O_7P  | 703.564 (H^+)     | 703.575     | 11       |
| PC(16:0/14:1)* | C_{38}H_{74}NO_8P | 704.523 (H^+)     | 704.524     | 1        |
|                |                   | 371.743 (H^+K)^2+ | 371.747    | 4        |
| PC(16:0/14:0)* | C_{38}H_{74}NO_8P | 716.530 (H^+)     | 716.541     | 11       |
| PC(O-32:1) or | C_{38}H_{74}NO_8P | 718.566 (H^+)     | 718.567     | 1        |
| PE(16:0/15:0)  | C_{38}H_{74}NO_8P | 720.554 (H^+)     | 720.557     | 3        |
| PE(O-36:6)     | C_{38}H_{74}NO_8P | 722.513 (H^+)     | 722.512     | -1       |
| PC(32:2) or PE(35:2) | C_{40}H_{79}O_7P  | 730.534 (H^+)     | 730.544     | 6        |
| PC(16:0/16:1)* | C_{40}H_{79}O_7P  | 732.554 (H^+)     | 732.553     | -1       |
|                |                   | 385.759 (H^+K)^2+ | 385.752    | -7       |
| PC(33:5) or PE(36:5) | C_{38}H_{78}NO_8P | 738.507 (H^+)     | 738.522     | 15       |
| PC(O-34:2))    | C_{38}H_{74}NO_8P | 744.591 (H^+)     | 744.587     | -4       |
| PC(O-34:1)     | C_{38}H_{74}NO_8P | 746.606 (H^+)     | 746.593     | -13      |
| PS(O-34:1)     | C_{38}H_{74}NO_8P | 748.541 (H^+)     | 748.542     | 1        |
| PC(34:4) or PE(37:4) | C_{42}H_{78}NO_8P | 754.531 (H^+)     | 754.536     | 5        |
| PC(16:0/18:3)* | C_{42}H_{78}NO_8P | 756.554 (H^+)     | 756.557     | 3        |
| PC(16:0/18:2)* | C_{42}H_{78}NO_8P | 758.570 (H^+)     | 758.570     | 0        |
| PC(16:0/18:1)* | C_{42}H_{78}NO_8P | 760.586 (H^+)     | 760.582     | -4       |
|                |                   | 399.775 (H^+K)^2+ | 399.778    | 3        |
| PC(16:0/18:0)* | C_{42}H_{78}NO_8P | 762.601 (H^+)     | 762.592     | -9       |
| PS(34:0) or    | C_{40}H_{78}NO_10P | 764.544 (H^+)     | 764.552     | 8        |
| PC(O-36:6)     | C_{42}H_{78}NO_8P | 764.559 (H^+)     | 764.552     | -7       |
| PC(35:5)       | C_{42}H_{78}NO_8P | 766.538 (H^+)     | 766.555     | 17       |
| PC(35:4)       | C_{42}H_{78}NO_8P | 768.554 (H^+)     | 768.561     | 7        |
| PC(16:0/20:6)* | C_{44}H_{80}NO_8P | 778.539 (H^+)     | 778.547     | 8        |
| PC(16:0/20:5)* | C_{44}H_{80}NO_8P | 780.553 (H^+)     | 780.551     | -2       |
|                |                   | 409.759 (H^+K)^2+ | 409.760    | 1        |
| PC(16:0/20:4)* | C_{44}H_{80}NO_8P | 782.570 (H^+)     | 782.572     | 2        |
| PC(16:0/20:3)* | C_{44}H_{80}NO_8P | 784.586 (H^+)     | 784.588     | 2        |
### Table 2. Cont.

| Lipid | Chemical formula | Monoisotopic mass | Measured mass | Δm (mDa) |
|-------|------------------|-------------------|---------------|----------|
| PC(16:0/20:2)* | C_{42}H_{84}NO_{10}P | 790.560 (H^+) | 790.568 | 8 |
| PC(O-38:7) | C_{45}H_{80}NO_{7}P | 802.537 (H^+) | 802.547 | 8 |
| PC(38:7) | C_{46}H_{78}NO_{8}P | 804.554 (H^+) | 804.558 | 4 |
| PC(16:0/22:6)* | C_{46}H_{80}NO_{10}P | 806.570 (H^+) | 806.569 | 1 |
| PS(16:0/22:6)* | C_{46}H_{80}NO_{10}P | 808.586 (H^+) | 808.587 | 1 |
| PS(38:4) | C_{44}H_{78}NO_{10}P | 812.544 (H^+) | 812.551 | 7 |
| PS(38:1) | C_{44}H_{84}NO_{10}P | 818.591 (H^+) | 818.595 | 4 |
| PS(O-38:5) | C_{44}H_{78}NO_{9}P | 822.565 (H^+) | 822.574 | 9 |
| PC(16:0/22:5)* | C_{46}H_{80}NO_{10}P | 824.580 (H^+) | 824.574 | 9 |
| PC(38:7) | C_{46}H_{78}NO_{8}P | 832.586 (H^+) | 832.585 | 4 |
| PC(16:0/22:5) | C_{46}H_{80}NO_{10}P | 838.560 (H^+) | 838.566 | 6 |
| PC(38:7) | C_{46}H_{78}NO_{8}P | 840.575 (H^+) | 840.584 | 9 |
| PC(16:0/22:6) | C_{46}H_{80}NO_{10}P | 840.573 (Na^+) | 840.584 | 11 |
| PS(38:1) | C_{46}H_{80}NO_{10}P | 845.533 (H^+) | 845.529 | 4 |
| PS(40:4) | C_{46}H_{82}NO_{10}P | 845.531 (Na^+) | 845.529 | 2 |
| PC(40:9) | C_{46}H_{80}NO_{10}P | 854.570 (H^+) | 854.561 | 9 |
| PC(40:7) | C_{46}H_{80}NO_{10}P | 875.713 (H^+) | 875.710 | 3 |
| PC(40:6) | C_{46}H_{80}NO_{10}P | 875.711 (Na^+) | 875.710 | 1 |
| PC(40:5) | C_{46}H_{80}NO_{10}P | 877.729 (H^+) | 877.725 | 4 |
| PC(40:4) | C_{46}H_{80}NO_{10}P | 881.757 (Na^+) | 881.755 | 2 |
| PC(40:3) | C_{46}H_{80}NO_{10}P | 883.775 (H^+) | 883.771 | 4 |
| PC(40:2) | C_{46}H_{80}NO_{10}P | 905.757 (Na^+) | 905.756 | 1 |
| TAG(54:8) | C_{57}H_{94}O_{6} | 875.713 (H^+) | 875.710 | 3 |
| TAG(52:5) | C_{57}H_{94}O_{6} | 877.729 (H^+) | 877.725 | 4 |
| TAG(54:7) | C_{57}H_{94}O_{6} | 881.757 (Na^+) | 881.755 | 2 |
| TAG(52:4) | C_{57}H_{94}O_{6} | 883.775 (H^+) | 883.771 | 4 |
| TAG(54:6) | C_{57}H_{94}O_{6} | 905.757 (Na^+) | 905.756 | 1 |
| TAG(52:3) | C_{57}H_{94}O_{6} | 907.744 (H^+) | 907.740 | 2 |
| TAG(54:5) | C_{57}H_{94}O_{6} | 918.742 (Na^+) | 918.740 | 2 |
| TAG(52:2) | C_{57}H_{94}O_{6} | 920.742 (Na^+) | 920.740 | 2 |
| TAG(54:4) | C_{57}H_{94}O_{6} | 932.742 (Na^+) | 932.740 | 2 |
| TAG(52:1) | C_{57}H_{94}O_{6} | 934.742 (Na^+) | 934.740 | 2 |
| PC(44:4) | C_{52}H_{80}NO_{8}P | 894.695 (H^+) | 894.708 | 13 |
| PC(44:3) | C_{52}H_{80}NO_{8}P | 896.711 (H^+) | 896.721 | 10 |
| PC(42:0) | C_{52}H_{80}NO_{8}P | 918.693 (Na^+) | 918.702 | 9 |
| PC(42:2) | C_{52}H_{80}NO_{8}P | 918.726 (H^+) | 918.733 | 7 |
| PC(44:1) | C_{52}H_{80}NO_{8}P | 900.742 (H^+) | 900.741 | 1 |
| TAG(56:9) | C_{56}H_{96}O_{6} | 901.729 (H^+) | 901.727 | 2 |
which showed fragments for the loss of oleic acid (18:1) and palmitic acid (16:0) at m/z 510.3631 and 484.3452, respectively (see Fig. 4b). Another major class of lipids in the spectra is LPC that is structurally similar to PC without the acyl group at sn-2. Phospholipase A2 is known to degrade PC into LPC and facilitate fertilization. Similar to PC, the major LPC is 16:0, and others include acyl chains between 14:0 and 22:6 (see Table 2). Because LPCs contain a free –OH group on the glycerol moiety, they undergo a loss of H2O during protonation and most ions are detected as [MH-H2O]+. These fragments are often more abundant than the protonated peaks. A few weak peaks corresponding to PE and PS can also be detected in the spectra. The major lipids found in the LAESI experiments and their fatty acid compositions are in good agreement with the literature reports based on extraction, HPLC separation and GC-based fatty acid analysis [12].

Table 2. Cont.

| Lipid       | Chemical formula | Monoisotopic mass | Measured mass | Δm (mDa) |
|-------------|------------------|-------------------|---------------|----------|
| TAG(58:8)   | C59H98O6         | 903.744 (H+)      | 903.741       | −3       |
|             |                  | 925.726 (Na+)     | 925.739       | 13       |
| TAG(56:7)   | C59H120O6        | 905.760 (H+)      | 905.754       | −6       |
|             |                  | 927.742 (Na+)     | 927.741       | −1       |
| TAG(56:6)   | C59H122O6        | 907.776 (H+)      | 907.771       | −5       |
|             |                  | 929.757 (Na+)     | 929.754       | −4       |
| TAG(54:3)   | C59H124O6        | 907.773 (Na+)     | 907.772       | −2       |
| TAG(56:5)   | C59H126O6        | 909.791 (H+)      | 909.788       | −3       |
| TAG(54:2)   | C59H128O6        | 909.789 (Na+)     | 909.788       | −1       |
| PC(46:6)    | C54H98NO8P       | 918.695 (H+)      | 918.702       | 7        |
| TAG(58:11)  | C61H98O6         | 925.727 (H+)      | 925.739       | 12       |
| TAG(58:10)  | C61H98O6         | 927.744 (H+)      | 927.741       | −3       |
|             |                  | 949.726 (Na+)     | 949.737       | 11       |
| TAG(58:9)   | C61H120O6        | 929.760 (H+)      | 929.754       | −6       |
|             |                  | 951.749 (Na+)     | 951.749       | 0        |
| PC(46:5)    | C59H27NO13       | 942.686 (Na+)     | 942.697       | 11       |
| PC(46:3)    | C54H102NO8P      | 946.724 (Na+)     | 946.726       | 2        |
| TAG(60:13)  | C63H98O6         | 949.729 (H+)      | 949.737       | 8        |
|             |                  | 971.711 (Na+)     | 971.714       | 3        |
| TAG(60:12)  | C63H98O6         | 951.744 (H+)      | 951.749       | 5        |
| TAG(60:11)  | C63H120O6        | 953.760 (H+)      | 953.757       | −3       |
|             |                  | 975.742 (Na+)     | 975.754       | 12       |
| TAG(58:8)   | C61H122O6        | 953.757 (Na+)     | 953.757       | 0        |
| TAG(60:10)  | C63H122O6        | 955.776 (H+)      | 955.765       | −11      |
| TAG(58:7)   | C63H124O6        | 955.773 (Na+)     | 955.765       | −8       |
| TAG(59:13)  | C63H126O6        | 957.695 (Na+)     | 957.688       | −7       |
| TAG(62:16)  | C65H94O6         | 971.713 (H+)      | 971.714       | 1        |
| TAG(62:14)  | C65H96O6         | 975.744 (H+)      | 975.754       | 10       |

*Lipids assignment based on tandem mass spectrometry.

doi:10.1371/journal.pone.0115173.t002
putative list of the assigned lipids observed in the LAESI mass spectra of de-jellied unfertilized X. laevis egg is shown in Table 2. LAESI mass spectra with annotations for selected FA, LPC and PC species are shown in Fig. 5.

Animal Pole vs. Vegetal Pole
The unfertilized X. laevis egg is polarized, with a highly pigmented animal pole and a weakly pigmented vegetal pole separated by an unpigmented equatorial zone. During oogenesis, mRNAs and proteins are synthesized and differentially stored in the cytoplasm of these regions for use by the embryo after fertilization. In particular, mRNAs that direct ectodermal development are enriched in the
animal pole whereas mRNAs that direct endoderm and gamete development are enriched in the vegetal pole \([40, 41]\). This molecular asymmetry is altered after fertilization by a cytoplasmic reorganization that occurs in response to sperm entry. This reorganization is critical for establishing the dorsal axis of the embryo. Because the mRNAs in the animal versus vegetal regions of the egg have very different developmental functions that are essential to establishing the vertebrate body plan, the corresponding metabolic and lipid profiles were investigated by LAESI-MS.

As seen in Fig. 6, numerous lipids were found in higher abundance at the vegetal pole compared to the animal pole. This finding was consistent with magnetic resonance imaging studies of the animal and vegetal poles in \(X. laevis\) oocytes that had shown two to three times higher abundance for TAGs in the vegetal region \([42]\). Among the many lipids detected at the vegetal pole, we found an LPC at \(m/z\) 478, which was consistent with the presence of yolk platelets there. A previous report suggested that LPC lipids played a role in assisting fertilization \([12]\). In addition, there were many small metabolites found at both poles and their putative identifications can be found in Table 1. Although differences in the small metabolite abundances at the two poles were clearly discernable, they exhibited significant variance between eggs. As an example, the ion abundance ratios between vegetal and animal poles are shown in Table 3 for certain metabolites and lipids. To establish reliable abundance ratios, further analysis is needed.

**Fertilized Ovum and Early Embryos**

A number of critical developmental events occur after the \(X. laevis\) egg is fertilized, including: sperm nucleus entry into the cytoplasm and an influx of calcium, the completion of meiosis II, the initiation of synchronous cell cycles, and the initiation of low levels of gene transcription at the 32-cell stage \([37]\). In the unfertilized egg, the cell cycle is arrested in metaphase of the second meiotic division, but all of the components for cell division, cytokinesis, gene transcription and protein synthesis have previously been produced in the oocyte.

We directly analyzed unperturbed eggs by LAESI-MS at 30 minute intervals after fertilization, and cleavage stage embryos at 8-cells and 32-cells. Most of the major metabolites and lipids found during these early stages after fertilization were similar to those in unfertilized \(Xenopus\) eggs (see Fig. 7). These results are consistent with the prevailing concept that the synthesis of the major cellular components occurs during oogenesis. However, some subtle differences between the some key metabolites were observed in the spectra. Most notably, spermine was consistently detected in the unfertilized eggs, but was absent in the 32-cell stage embryos. It is well documented that the concentration of spermine decreases during oogenesis leading to the reduction of total polyamines required for maturation \([43, 44]\).

In both jellied and dejellied systems, the intensities for the following ions from stage 8 embryos were consistently lower than from unfertilized eggs: \(m/z\) 113.884, 114.892, 150.009, 175.118 (arginine), 199.991, 231.934, 239.919 and 277.199.
Fig. 5. Positive LAESI mass spectra of unfertilized *X. laevis* egg with annotations for selected (a) fatty acids (FA), (b) lysophosphatidylcholines (LPC), and (c) phosphatidylcholines (PC). A complete list of the detected lipid ions is presented in Table 2.

doi:10.1371/journal.pone.0115173.g005
Conversely, the following ions showed higher abundances in stage 8 embryos: 113.104, 131.111 (acetylputrescine), 157.127, 261.075, 239.101, 277.054, 283.066, 299.038, 315.015, 239.236, 241.098, 284.066, 295.021, 542.052 and 660.618 (Cer(d18:0/23:0)). Furthermore, there is a higher abundance of most TAGs in stage 8 embryos compared to unfertilized dejellied *Xenopus* eggs.

**Conclusions**

There has been extensive of the transcriptome that regulates the major developmental transitions in *X. laevis* at the subcellular level. These investigations have been limited to the detection of mRNAs by *in situ* hybridization and RT-PCR approaches and to the detection of proteins by immunohistochemistry or Western blotting techniques. The latter have been hampered by the lack of antibodies against proteins of interest. In addition, these approaches reliably detect mRNAs and proteins that are in high abundance. Metabolites and lipids have not been studied in detail at in single eggs because of the lack of appropriately sensitive technologies that can assess these molecules in the living egg/embryo in its natural state. Therefore, evaluation of eggs and embryos with LAESI-MS was performed to determine whether subtle changes in metabolites and lipids and their subcellular profiles are detectable.

In this report the direct atmospheric pressure analysis of *X. laevis* eggs and early cleavage stages embryos is presented. The results of the LAESI-MS metabolic and lipidomic profiles of the *Xenopus* eggs were compared to previous literature results and showed a wide coverage for lipids and small metabolites. Reactive-LAESI MS/MS, with in-plume cationization of PC lipids without sample preparation, was
### Table 3. Metabolite and lipid ion abundance ratios for vegetal and animal poles of *X. laevis* egg.

| Metabolite or Lipid                      | Ion     | m/z      | Abundance ratio vegetal/animal | Abundance ratio animal/vegetal |
|-----------------------------------------|---------|----------|--------------------------------|---------------------------------|
| choline                                 | [+]     | 104.1    | 3.7                            | 0.3                             |
| adenine                                 | [H+]    | 136.1    | 1.0                            | 1.0                             |
| guanine or hydroxyadenine               | [Na+]   | 174.0    | 4.9                            | 0.2                             |
| arginine                                | [H+]    | 175.1    | 1.8                            | 0.6                             |
| phosphocholine                          | [H+]    | 184.1    | 5.2                            | 0.2                             |
| monosaccharide                          | [Na+]   | 203.1    | 1.2                            | 0.9                             |
| acetylhydrolipoamide                    | [H+]    | 250.1    | 2.8                            | 0.4                             |
| glutathione                             | [Na+]   | 330.1    | 0.6                            | 1.7                             |
| glutathione                             | [K+]    | 346.0    | 0.7                            | 1.5                             |
| disaccharide or trehalose               | [Na+]   | 365.1    | 1.9                            | 0.5                             |
| disaccharide or trehalose               | [K+]    | 381.1    | 2.8                            | 0.4                             |
| unknown m/z 374.033                     | -       | 374.0    | 7.5                            | 0.1                             |
| cholesterol                             | [H⁺⁻H₂O]| 369.4    | 4.7                            | 0.2                             |
| adenosine diphosphate (ADP)             | [H⁺]    | 428.0    | 0.5                            | 1.9                             |
| adenosine diphosphate (ADP)             | [Na⁺]   | 450.0    | 1.7                            | 0.6                             |
| unknown m/z 467.101                     | -       | 467.1    | 0.1                            | 17.0                            |
| LPC(16:1)                               | [H⁺⁻H₂O]| 476.3    | 4.1                            | 0.2                             |
| LPC(16:0)                               | [H⁺⁻H₂O]| 478.3    | 4.8                            | 0.2                             |
| adenosine triphosphate (ATP)            | [H⁺]    | 508.0    | 1.4                            | 0.7                             |
| adenosine triphosphate (ATP)            | [Na⁺]   | 530.0    | 0.6                            | 1.8                             |
| unknown m/z 522.590                     | -       | 522.6    | 1.3                            | 0.7                             |
| trisaccharides                          | [Na⁺]   | 527.2    | 2.4                            | 0.4                             |
| unknown m/z 541.126                     | -       | 541.1    | 1.2                            | 0.8                             |
| Cer(d 40:1)                             | [Na⁺]   | 660.6    | 0.1                            | 11.6                            |
| unknown m/z 689.201                     | -       | 689.2    | 2.0                            | 0.5                             |
| PC(16:0/16:1)                           | [H⁺]    | 732.6    | 5.6                            | 0.2                             |
| PC(34:4) or PE(37:4)                    | [H⁺]    | 754.5    | 9.2                            | 0.1                             |
| PC(16:0/18:3)                           | [H⁺]    | 756.6    | 3.9                            | 0.3                             |
| PC(16:0/18:2)                           | [H⁺]    | 758.6    | 3.7                            | 0.3                             |
| PC(16:0/18:1)                           | [H⁺]    | 760.6    | 4.3                            | 0.2                             |
| PC(16:0/18:0)                           | [H⁺]    | 762.6    | 3.3                            | 0.3                             |
| PC(16:0/20:5)                           | [H⁺]    | 780.6    | 5.9                            | 0.2                             |
| PC(16:0/20:4)                           | [H⁺]    | 782.6    | 6.6                            | 0.2                             |
| PC(16:0/20:3)                           | [H⁺]    | 784.6    | 3.7                            | 0.3                             |
| PS(O-38:5)                              | [H⁺]    | 796.5    | 3.0                            | 0.3                             |
| PC(16:0/22:6)                           | [H⁺]    | 806.6    | 5.8                            | 0.2                             |
| PC(16:0/22:5)                           | [H⁺]    | 808.6    | 5.5                            | 0.2                             |
| TAG(54:6)+[H⁺] or TAG(52:3)+[Na⁺]        | [H⁺] or | 879.7    | 18.0                           | 0.1                             |
| TAG(52:3)                               | [K⁺]    | 895.1    | 9.9                            | 0.1                             |
| TAG(52:2)                               | [K⁺]    | 897.7    | 9.4                            | 0.1                             |
| TAG(56:9)                               | [H⁺]    | 901.7    | 9.2                            | 0.1                             |
| TAG(56:8)                               | [H⁺]    | 903.7    | 16.8                           | 0.1                             |
| TAG(54:6)                               | [K⁺]    | 917.7    | 9.8                            | 0.1                             |
utilized to identify the acyl chains. An investigation of the animal and vegetal poles of the X. laevis ovum showed increased abundance of lipids in the vegetal pole relative to the animal pole. Radial profiling of a jelly coated egg by LAESI-MS utilizing consecutive laser pulses revealed dramatic compositional changes between the jelly coat the vitelline/plasma membranes and the cytoplasm.

Finally, LAESI-MS was used to demonstrate that subtle metabolic profile changes could be detected after the egg was fertilized and synchronous cell cycles were initiated. Of particular note, spermine was depleted after fertilization, which consistent with the reduction of total polyamines required for embryo maturation [43, 44]. Major differences between the metabolites of unfertilized egg and the early cleavage stage embryos were not expected because all the needed organelles and biochemical building blocks needed to carry the embryo through the first 8 hours of development had previously been synthesized and stored in the oocyte. Nonetheless, the sensitivity of LAESI-MS enabled the detection of a few subtle differences, and uncovered novel molecules that are asymmetrically distributed between the animal, ectoderm-forming, and vegetal, endoderm-forming, regions of the egg. This simple approach, partially validated here for a vertebrate embryo

| Metabolite or Lipid | Ion  | m/z  | Abundance ratio vegetal/animal | Abundance ratio animal/vegetal |
|---------------------|------|------|-------------------------------|-------------------------------|
| TAG(54:5)           | [K⁺] | 919.7| 7.3                           | 0.1                           |
| TAG(54:4)           | [K⁺] | 921.7| 6.7                           | 0.1                           |
| TAG(60:13)          | [H⁺] | 949.7| 7.3                           | 0.1                           |

Fig. 7. Positive ion LAESI mass spectra of an unfertilized Xenopus ovum and a 32-cell embryo show a very similar metabolite profiles with some key differences, e.g., the absence of spermine in the embryo spectrum. The inset shows a Xenopus embryo at the 32-cell stage. The scale bar in the inset is 200 μm.

doi:10.1371/journal.pone.0115173.g007

doi:10.1371/journal.pone.0115173.t003
for which there is abundant traditional biochemical data, should be ideal for the
direct analyses of other embryos that are rare and smaller in size.

Acknowledgments
One of the authors (P. S.) thanks the Director of the Indian Institute of Chemical
Technology, Hyderabad, and the Council of Scientific and Industrial Research,
New Delhi, India for granting leave.

Author Contributions
Conceived and designed the experiments: AV SAM. Performed the experiments:
BS PS. Analyzed the data: BS PS BRS HDH MJP SAM AV. Contributed reagents/
materials/analysis tools: SAM. Wrote the paper: BS PS BRS HDH MJP SAM AV.

References

1. O’Farrell PH, Stumpff J, Tin Su T (2004) Embryonic Cleavage Cycles: How Is a Mouse Like a Fly?
   Current biology: CB 14: R35–R45.
2. Vastag L, Jorgensen P, Peshkin L, Wei R, Rabinowitz JD, et al. (2011) Remodeling of the
   Metabolome during Early Frog Development. PLoS ONE 6: e16891.
3. Gurdon JB, Hopwood N (2000) The introduction of Xenopus laevis into developmental biology: Of
   empire, pregnancy testing and ribosomal genes. International Journal of Developmental Biology 44: 43–
   50.
4. Sato K-i, Yoshino K-i, Tokmakov AA, Iwasaki T, Yonezawa K, et al. (2006) Studying Fertilization in
   Cell-Free Extracts Xenopus Protocols. In: Liu XJ, editor: Humana Press. pp. 395–411.
5. Gurdon JB, Hopwood N (2000) The introduction of Xenopus laevis into developmental biology: of
   empire, pregnancy testing and ribosomal genes. International Journal of Developmental Biology 44: 43–
   50.
6. Khokha MK (2012) Xenopus white papers and resources: Folding functional genomics and genetics into
   the frog. Genesis 50: 133–142.
7. Elkan ER (1938) The Xenopus Pregnancy Test. BMJ 2: 1253–1274.
8. Wasserman WJ, Masui Y (1976) A cytoplasmic factor promoting oocyte maturation: its extraction and
   preliminary characterization. Science 191: 1266–1268.
9. Murray AW, Kirschner MW (1989) Cyclin Synthesis Drives The Early Embryonic-cell Cycle. Nature 339:
   275–280.
10. Stith BJ, Hall J, Ayres P, Waggoner L, Moore JD, et al. (2000) Quantification of major classes of
    Xenopus phospholipids by high performance liquid chromatography with evaporative light scattering
    detection. Journal of Lipid Research 41: 1448–1454.
11. Zhang J, Xie Y, Hedrick JL, Lebrilla CB (2004) Profiling the morphological distribution of O-linked
    oligosaccharides. Analytical Biochemistry 334: 20–35.
12. Petcoff DW, Holland WL, Stith BJ (2008) Lipid levels in sperm, eggs, and during fertilization in
    Xenopus laevis. Journal of Lipid Research 49: 2365–2378.
13. Fletcher JS, Lockyer NP, Vaidyanathan S, Wickerman JC (2007) TOF-SIMS 3D Biomolecular Imaging
    of Xenopus laevis Oocytes Using Buckminsterfullerene (C60) Primary Ions. Analytical Chemistry 79:
    2199–2206.
14. Koek MM, Bakker F, Engel W, van den Maagdenberg A, Ferrari MD, et al. (2009) Metabolic Profiling
    of Ultrasmall Sample Volumes with GC/MS: From Microliter to Nanoliter Samples. Analytical Chemistry
    82: 156–162.
15. Shechter D, Nicklay JJ, Chitta RK, Shabanowitz J, Hunt DF, et al. (2009) Analysis of histones in Xenopus laevis I. a distinct index of enriched variants and modifications exists in each cell type and is remodelled during developmental transitions. Journal of Biological Chemistry 284: 1064–1074.

16. Yurewicz EC, Oliphant G, Hedrick JL (1975) Macromolecular composition of Xenopus laevis egg jelly coat. Biochemistry 14: 3101–3107.

17. Ferreira CR, Eberlin LS, Hallett JE, Cooks RG (2012) Single oocyte and single embryo lipid analysis by desorption electrospray ionization mass spectrometry. Journal of Mass Spectrometry 47: 29–33.

18. Gonzalez-Serrano AF, Pirro V, Ferreira CR, Oliveri P, Eberlin LS, et al. (2013) Desorption Electrospray Ionization Mass Spectrometry Reveals Lipid Metabolism of Individual Oocytes and Embryos. Plos One 8.

19. Nemes P, Vertes A (2007) Laser ablation electrospray ionization for atmospheric pressure, in vivo, and imaging mass spectrometry. Anal Chem 79: 8098–8106.

20. Nemes P, Barton AA, Li Y, Vertes A (2008) Ambient Molecular Imaging and Depth Profiling of Live Tissue by Infrared Laser Ablation Electrospray Ionization Mass Spectrometry. Analytical Chemistry 80: 4575–4582.

21. Shrestha B, Nemes P, Vertes A (2010) Ablation and analysis of small cell populations and single cells by consecutive laser pulses. Applied Physics A: Materials Science & Processing 101: 121–126.

22. Stolee JA, Shrestha B, Mengistu G, Vertes A (2012) Observation of Subcellular Metabolite Gradients in Single Cells by Laser Ablation Electrospray Ionization Mass Spectrometry. Angewandte Chemie International Edition 51: 10386–10389.

23. Sripadi P, Shrestha B, Easley RL, Carpio L, Kehn-Hall K, et al. (2010) Direct detection of diverse metabolic changes in virally transformed and tax-expressing cells by mass spectrometry. PLoS One 5: e12590.

24. Shrestha B, Vertes A (2009) In situ metabolic profiling of single cells by laser ablation electrospray ionization mass spectrometry. Analytical Chemistry 81: 8265–8271.

25. Moody SA (1999) Cell Lineage Analysis in Xenopus Embryos. pp. 331–347.

26. Sive HL, Grainger RM, Harland RM (2000) Early Development of Xenopus laevis. A Laboratory Manual. Cold Spring Harbor, NY: Cold Spring Harbor Laboratory Press. 338 p.

27. Shrestha B, Vertes A (2014) High-Throughput Cell and Tissue Analysis with Enhanced Molecular Coverage by Laser Ablation Electrospray Ionization Mass Spectrometry Using Ion Mobility Separation. Analytical Chemistry 86: 4308–4315.

28. Smith CA, Maille GO, Want EJ, Qin C, Trauger SA, et al. (2005) METLIN: A Metabolite Mass Spectral Database. Therapeutic Drug Monitoring 27: 747–751.

29. Caspi R, Foerster H, Fulcher CA, Kaipa P, Krummenacker M, et al. (2008) The MetaCyc Database of metabolic pathways and enzymes and the BioCyc collection of Pathway/Genome Databases. Nucleic Acids Research 36: D623–D631.

30. Fahy E, Sud M, Cotter D, Subramaniam S (2007) LIPID MAPS online tools for lipid research. Nucleic Acids Research 35: W606–W612.

31. Wishart DS, Tzur D, Knox C, Eisner R, Guo AC, et al. (2007) HMDB: the Human Metabolome Database. Nucleic Acids Research 35: D521–D526.

32. Kanehisa M, Goto S (2000) KEGG: Kyoto Encyclopedia of Genes and Genomes. Nucleic Acids Research 28: 27–30.

33. Kanehisa M, Goto S, Sato Y, Furumichi M, Tanabe M (2012) KEGG for integration and interpretation of large-scale molecular data sets. Nucleic Acids Research 40: D109–D114.

34. Hannappel E, Kalbacher H, Voelter W (1988) Thymosin J1Xen4: A new thymosin J4-like peptide in oocytes of Xenopus laevis. Archives of Biochemistry and Biophysics 260: 546–551.

35. Olson JH, Xiang X, Ziegert T, Kittelson A, Rawls A, et al. (2001) Allurin, a 21-kDa sperm chemoattractant from Xenopus egg jelly, is related to mammalian sperm-binding proteins. Proceedings of the National Academy of Sciences 98: 11205–11210.
36. Xiang X, Kittelson A, Olson J, Bieber A, Chandler D (2005) Allurin, a 21 kD sperm chemoattractant, is rapidly released from the outermost jelly layer of the Xenopus egg by diffusion and medium convection. Molecular Reproduction and Development 70: 344–360.

37. Hausen P, Riebesell M (1991) The Early Development of Xenopus Laevis: An Atlas of the Histology. Berlin, Heidelberg, New York: Springer-Verlag. 6 p.

38. Ohlendorf DH, Barbarash GR, Trout A, Kent C, Banaszak LJ (1977) Lipid and polypeptide components of the crystalline yolk system from Xenopus laevis. Journal of Biological Chemistry 252: 7992–8001.

39. Shrestha B, Nemes P, Nazarian J, Hathout Y, Hoffman EP, et al. (2010) Direct analysis of lipids and small metabolites in mouse brain tissue by AP IR-MALDI and reactive LAESI mass spectrometry. Analyst 135: 751–758.

40. Heasman J (2006) Patterning the early Xenopus embryo. Development 133: 1205–1217.

41. King ML, Messitt TJ, Mowry KL (2005) Putting RNAs in the right place at the right time: RNA localization in the frog oocyte. Biology of the Cell 97: 19–33.

42. Lee S-C, Cho J-H, Mietchen D, Kim Y-S, Hong KS, et al. (2006) Subcellular In Vivo 1H MR Spectroscopy of Xenopus laevis Oocytes. Biophysical Journal 90: 1797–1803.

43. Osborne HB, Mulner-Lorillon O, Marot J, Belle R (1989) Polyamine levels during Xenopus laevis oogenesis: A role in oocyte competence to meiotic resumption. Biochemical and Biophysical Research Communications 158: 520–526.

44. Bassez T, Paris J, Omilli F, Dorel C, Osborne HB (1990) Post-transcriptional regulation of ornithine decarboxylase in Xenopus laevis oocytes. Development 110: 955–962.