Anomalous Spectral weight in photoemission spectra of the hole doped Haldane chain
\(Y_{2-x}Sr_xBaNiO_5\).

Y. Fagot-Revurat and D. Malterre
Laboratoire de Physique des Matériaux,
Université Henri Poincaré, Nancy I - B.P. 239
F-54506 Vandœuvre-lès-Nancy, France

F.-X. Lannuzel, E. Janod and C. Payen
Institut des Matériaux Jean Rouxel,
Université de Nantes-CNRS,
F-44322 Nantes Cedex 3, France

L. Gavioli and F. Bertran
Laboratoire pour l’Utilisation du Rayonnement Electromagnétique,
Université Paris-Sud
F-91405 Orsay Cedex, France

(Dated: November 11, 2018)

In this paper, we present photoemission experiments on the hole doped Haldane chain compound \(Y_{2-x}Sr_xBaNiO_5\). By using the photon energy dependence of the photoemission cross section, we identified the symmetry of the first ionisation states (d type). Hole doping in this system leads to a significant increase in the spectral weight at the top of the valence band without any change in the vicinity of the Fermi energy. This behavior, not observed in other charge transfer oxides at low doping level, could result from the inhomogeneous character of the doped system and from a Ni 3d-O 2p hybridization enhancement due to the shortening of the relevant Ni-O distance in the localized hole-doped regions.

PACS numbers: 71.20.-b, 71.27.+a, 71.30.+h

I. INTRODUCTION

Oxides of 3d transition metals have attracted considerable attention in the last two decades because they exhibit very varied and fascinating properties (metal-insulator transition, high Tc superconductivity, intriguing magnetic properties). These behaviors partly result from an interplay of low dimensionality and electronic correlations. Many transition metal oxides exhibit a two dimensional electronic structure. Due to its peculiar crystal structure, \(Y_2BaNiO_5\) is essentially a one dimensional (1D) divalent nickel oxide. It crystallizes in a body-centered orthorhombic structure within the Immm space group. The structure is characterized by linear chains of NiO_6 octahedra sharing corners along the z direction. The octahedra are compressed along the chains resulting in two short Ni-O distances (0.188 nm) and four longer Ni-O distances (0.219 nm). This unusual contraction and vanishing interactions between neighboring chains lead to a quasi-one-dimensional electronic structure. The hybridization between Ni-3d and O-2p states is strong along the chain direction. Yet, \(Y_2BaNiO_5\) has an insulating ground state due to exchange and correlation effects. The superexchange antiferromagnetic coupling between neighboring S=1 spins carried by Ni^{2+} ions in \(Y_2BaNiO_5\) yields a quantum spin liquid ground state. Large quantum spin fluctuations associated with the 1D character of the Ni-O-Ni network prevent the formation of an ordered magnetic state. The spin liquid can be described as macroscopically coherent quantum states with a Haldane gap in the spin excitation spectrum. Each S=1 spin can be considered as the triplet state formed from the two ferromagnetically coupled S=1/2 holes of Ni^{2+}. Two S=1/2 holes of neighboring sites form a singlet state. The ground state is then described in the well-known valence bond solid (VBS) picture. Excitations are triplet states obtained by breaking valence bonds. Hole doping in this spin liquid state results in interesting behaviors. Low-energy spin excitations are characterized by incommensurate spin density modulation around impurities, and the dynamical spin structure factor reveals spectral weight in the Haldane gap.

\(\bar{Y}\)From an electronic point of view, \(Y_2BaNiO_5\), like most of the Ni and Cu oxides, belongs to the charge transfer regime in the Zaannen-Sawatzky-Allen diagram. Then if the low-energy properties usually associated with spin excitations could be very different, the charge excitations and in particular the effect of doping should look very similar in these families. The photoemission spectrum of homogeneously hole doped systems only exhibits small spectral weight modifications with doping whereas large spectral weight transfer is observed in the inverse photoemission spectrum. In charge transfer oxides, a p symmetry is expected for the first ionization states. Thus, the substitution of Sr or Ca divalent ions...
for trivalent Y in $Y_{2-x}Sr_xBaNiO_5$ should yield holes in the ligand p band. Nevertheless, hybridization between Ni-3d states and O-2p states can modify this picture by shifting highly hybridized $d^\pi L$ states toward the top of the O-2p band ($L$ represents a hole on the ligand). As a consequence, the first ionization states probed by photoemission should have a 3d character. This spectroscopy, by measuring the electron-removal spectrum, can give information about the symmetry of the first ionization states and about the modification of electronic properties in hole doped compounds. The electronic structure of hole doped $Y_2BaNiO_5$ has been investigated with several methods including x-ray absorption spectroscopy (XAS) optical conductivity, photoemission and inverse photoemission. These experiments show that doped holes have either $O(2p_z)$ or mixed $Ni3d_{z^2-r^2} - O(2p_z)$ character. Hole doping introduces new empty localized states in the charge gap region without any spectral weight very close to the Fermi level. The doped carriers are trapped in local bound states and should therefore locally distort the electronic structure only in the vicinity of their location giving rise to a possible non-homogeneous doping at the nanoscopic scale. Therefore, the system does not become metallic upon doping and no charge ordering is observed.

In this paper, we investigate the electronic structure of $Y_{2-x}Sr_xBaNiO_5$ for $x = 0, 0.1$ and $0.2$ by photoemission spectroscopy. By using the photon energy dependence of the spectral density, we confirm that the first electron removal excitations have mainly a Ni-3d character. Moreover, these measurements illustrate the effect of hole doping in this system by evidencing an increase in the spectral weight of d symmetry at the top of the valence band. In contrast to most of hole doped cuprates and nickelates which only present significant spectral modifications for the unoccupied electronic states, a large spectral weight modification is observed in the photoemission spectra of $Y_{2-x}Sr_xBaNiO_5$ for doping level as low as $x = 0.1$. This singular behavior probably results from the large $Ni3d - O2p$ hybridization along the chain direction and from the inhomogeneous character of the doped system.

II. EXPERIMENTAL DETAILS

Polycrystalline samples of $Y_{2-x}Sr_xBaNiO_5$ ($x = 0.0, 0.10$ and $0.20$) were prepared by standard solid state reactions and characterized by X-ray powder diffraction, microprobe and thermogravimetric analysis. Samples are single-phased, homogeneously doped and stoichiometric in oxygen. Photoemission experiments have been carried out on the SU3 beamline of the French synchrotron facilities (LURE). The photon energy used was in the range extending from 60 eV to 200 eV. The sample surfaces were scraped in situ with a diamond file to obtain clean surfaces under the ultra high vacuum of low $10^{-10}$ Torr. The cleanliness of the surfaces was checked by the lack of contaminant in the valence band spectrum and by X-ray photoemission core level spectra. The Fermi energy was calibrated by measuring the metallic edge of a Cu foil close to the sample. All measurements were carried out at room temperature with an energy resolution better than 50 meV.

III. RESULTS AND DISCUSSION

In figure 1, we present angle-integrated photoemission spectra of $Y_2BaNiO_5$ for photon energy varying between 60 eV and 200 eV. These spectra exhibit three features at $E \approx -2.0$ eV, $E \approx -3.5$ eV and $E \approx -5.0$ eV, hereafter referred to as A, B and C, respectively. The peak at $-10$ eV, usually observed in Ni oxides could originate from multielectronic effects and/or impurity features. The spectra are arbitrarily normalized on the C structure. With this normalization, a very strong photon energy dependence of the A feature intensity is clearly observed. Owing to the different photon energy dependence of d and p states cross section, this behavior suggests that the A feature is dominated by the Ni-3d states whereas the B and C feature have likely p character. This result is in complete agreement with the previous measurements carried out in the Ca substituted oxide ($Y_{2-x}Ca_xBaNiO_5$) with HgI$_2$ ($h\nu=21.2$ eV) and Mg-$K\alpha(h\nu=1253.6$ eV) excitations which exhibit the same trend. Moreover, spin polarized band calculations in the local density approximation confirm this statement by showing that the C feature can be associated with O2p-Ni3d bonding states with dominant oxygen character whereas the states in the energy region of the B feature correspond to oxygen 2p-2p interactions with a non-bonding character with respect to Ni3d-O2p interactions. A satisfactory agreement between band calculation and photoemission spectra are

![FIG. 1: (a) Angle-integrated photoemission spectra of $Y_2BaNiO_5$ for several photon energies in the 60 eV-200 eV range. The spectra are arbitrary normalized on the maximum of the C structure. (b) $h\nu=190$ eV- spectrum of NiO.](image-url)
found for these two kinds of states. The top of the valence band in the calculations corresponds to antibonding states with a dominant Ni-3d character. Recent LSDA+U calculations, which predict a quite correct energy gap value, point out the highly hybridized character of these states. A significant disagreement between band calculated density of states and photoemission spectrum is exhibited for these states. This is not surprising since the A structure represents excitations of correlated d states which are poorly described by one-electron (ground state) approaches. Such a valence spectrum is usually observed in divalent Ni oxides like NiO, as observed in Fig.1(b) where the photoemission spectrum of NiO recorded with 190 eV photon-energy is presented. In NiO, the main structure close to -2eV has been interpreted to be due to $d^9$ states. The Ni-3d derived structures are less resolved for $Y_2BaNiO_5$ than for NiO (at the same photon energy) because the transition metal to oxygen atomic ratio is five time larger in the $Y_2BaNiO_5$ compound. However, an increase in the absolute intensity of this feature can be evidenced by comparison with figure 1 shows that this feature has a Ni-3d character. Comparison with figure 1 shows that the main effect of hole doping is to increase the intensity of feature A. Moreover, a narrowing of the bandwidth is also observed. In order to illustrate the increase in intensity, we report in figure 3 the photoemission spectra of $Y_{1-x}Sr_{x}BaNiO_5$ for several photon energies in the 60 eV-200 eV range.

FIG. 2: Angle-integrated photoemission spectra of $Y_{1-x}Sr_{x}BaNiO_5$ for several photon energies in the 60 eV-200 eV range.

FIG. 3: Angle-integrated Photoemission spectra for several photon energies in the 60 eV-200 eV range.
oxides remain insulating. Secondly, even for low x, the A feature exhibits a very large increase in intensity with doping. This behavior contrasts to the usually observed spectral weight modification in hole doped divalent Ni oxides, and more generally in cuprates and nickelates. In $La_{2-x}Sr_xNiO_4$, for instance, no significant increase in the spectral weight intensity is observed for doping level smaller than $x=0.5$. Unfortunately, due to the relative content of O-2p and Ni-3d states, it is not really possible to quantitatively estimate the weight of the $d^{x\downarrow}_L$ state. Transfers of spectral weight in hole-doped strongly correlated systems have been previously discussed in detail in the literature. Anomalous spectral weight modifications related to electronic correlations are expected in the electron-addition part (unoccupied states) of hybridized charge transfer systems. This trend is indeed observed in the XAS oxygen K edge of $Y_{2−x}(Ca, Sr)_xBaNiO_5$ as well as in inverse photoemission. However the electron-removal part of the spectrum (occupied states) should slightly decrease upon doping within this approach (since some low energy states are depopulated) in contrast to the spectroscopic behavior. An additional mechanism has to be invoked to explain the evolution of photoemission spectra with Sr content. Interestingly, hole doping in $Y_2BaNiO_5$ could result in an increase in the Ni3d-O2p hybridization since, unlike in $La_{2-x}Sr_xNiO_4$, the average Ni-O distance shortens continuously upon doping (see inset figure 3 for $Y_{2−x}Sr_xBaNiO_4$ and references [55,11] for $Y_{2−x}Ca_xBaNiO_5$). As the apical Ni-O distance in $Y_2BaNiO_5$ is the shortest one observed in nickelates, a small decrease should lead to a large increase in hybridization. This will be emphasized in case of an inhomogeneous doping. If the doped holes are trapped in local bound states, disturbing the electronic states only in their vicinity, the real Ni-O bonds in these perturbed part of the sample should be shorter than the average one measured by X-rays. As a consequence, in the photoemission spectra, which are the simple ad-
relevant Ni and O orbitals, gives large values of transfer integrals within a hole-doped NiO$_6$ cluster.$^{21}$ This singular spectroscopic signature, only observed in doped Y$_2$BaNiO$_5$, could result from the particularly small Ni-O distance characterizing this quasi-1D compound.

IV. CONCLUSION

To summarize, we have performed an angle-integrated photoemission investigation of the spectral weight modification by hole doping in the Y$_{2-x}$Sr$_x$BaNiO$_5$ oxides. In this Haldane chain compound, hole doping leads to strong modifications in the unoccupied part of the spectral density as usually observed in cuprates and nickelates. However, photoemission spectroscopy exhibits a singular behavior since a significant spectral weight increase corresponding to final states with d character is observed. We propose that this surprising behavior could result from an increase in hybridization between Ni-3d and O-2p states induced by the reduction of Ni-O distance in the chain direction by doping. Such an effect is favoured by the inhomogeneous character of doping. We hope that reliable calculations will be performed in the future to understand this anomalous spectral weight enhancement.

V. ACKNOWLEDGEMENT

We would like to thank M. Grioni for helpful discussion and A. Taleb for her hospitality on the SU3 beamline.

1. J. Amador, E. Gutierrez-Puebla, M.A. Monge, I. Rasines, C. Ruiz-Valero, F. Fernandez, R. Saez-Puche and J.A. Campa, Phys. Rev. B 42, 7918 (1990).
2. L.F. Mattheiss, Phys. Rev. B 48, 4352 (1993).
3. J. Darriet and L.P. Regnault, Solid State Commun. 856, 409 (1993).
4. I. Affleck, T. Kennedy, E.H. Lieb and H. Tasaki, Phys. Rev. Lett. 59, 799 (1987).
5. J.F. Ditusa, S.W. Cheong, J.H. Park, G. Aeppli, C. Broholm, and C.T. Chen, Phys. Rev. Lett. 73, 1857 (1994).
6. E. Dagotto, J. Riera, A. Sandvik and A. Moreo, Phys. Rev. B 76, 1731 (1996).
7. J.G. Xu, G. Aeppli, M.E. Bisher, C. Broholm, J.F. Ditusa, C.D. Frost, T. Ito, K. Oka, R.L. Paul, H. Takagi and M.M.J. Treacy, Science 289 (2000).
8. J. Zaanen, G.A. Sawatzky, and J.W. Allen, Phys. Rev. Lett. 55, 418 (1985).
9. I. Ito, H. Yamaguchi, K. Oka, K.M. Kojima, H. Eisaki and S. Uchida, Phys. Rev. B 64, 064001(R) (2001).
10. K. Maiti, P. Mahadevan and D.D. Sarma, Phys. Rev. B 59, 12457 (1999).
11. F.X. Lannuzel, E. Janod, C. Payen, G. Ouvrard, P. Moreau, O. Chauvet, P. Parent, C. Laffon, J. Alloys Comp. 317-318, 149 (2001).
12. K. Maiti and D.D. Sarma, Phys. Rev. B 58, 9746 (1998).
13. P. Novác F. Boucher, P. Grossier, P. Blaha, K. Schwarz Phys. Rev. B 63, 235114 (2001).
14. A. Fujimori and F. Minami, Phys. Rev. B 30, 957 (1984).
15. J. Zaanen and G.A. Sawatzky, J. Solid State Chem. 88, 8 (1990); Prog. Theor. Phys. Suppl. 101, 231 (1990).
16. H.Eskes, M.B. J. Meinders and G.A. Sawatzky, Phys. Rev. Lett. 67, 1035 (1991).
17. H. Eisaki, S. Uchida, T. Mizokawa, H. Namatame, A. Fujimori, J. van Elp, P. Kuiper, G.A. Sawatzky, S. Hosoya and H. Katayama-Yoshida, Phys. Rev. B 45, 12513 (1992).
18. P.J. Heaney, A. Mehta, G. Sarosi, V.E. Lambert, A. Navrotsky, Phys. Rev. B 57, 10370 (1998). In this work, a careful analysis of the position of the oxygen in the plane of La$_{2-x}$Sr$_x$NiO$_4$ indicates that despite the shortening of the cell parameters, the Ni-O distance may even increase due to a rotation of the NiO$_6$ octahedra.
19. J.A. Alonso, I. Rasines, J. Rodriguez-Carvajal and J.B. Torrance, J. Solid State Chem., 109, 231 (1994).
20. V. Massaroti, D. Capsoni, M. Bini, A. Altomare and A.G. Moliterni, Z. Kristallogr. 214, 231 (1999).
21. C.D. Batista, A.A. Aliglia and J. Eroles, Europhys. Lett. 43, 71 (1998).