Preparation and Photo luminescent Properties of \( \text{Sr}_2\text{Si}_5\text{N}_8: \text{Eu}^{2+} \) Red Phosphors for White Light-Emitting Diodes

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Abstract. This paper study a simple, inexpensive synthetic route to \( \text{Sr}_2\text{Si}_5\text{N}_8: \text{Eu}^{2+} \) based red nitride phosphors for white light-emitting diodes (WLEDs). The \( \text{Eu}^{2+} \) doped \( \text{Sr}_2\text{Si}_5\text{N}_8 \) nitride phosphor were synthesized by carbothermal reduction and nitridation method using \( \text{SrCO}_3, \text{Si}_3\text{N}_4 \) and \( \text{Eu}_2\text{O}_3 \) as raw materials, and carbon as main reduction agent. \( \text{Sr}_2\text{Si}_5\text{N}_8: \text{Eu}^{2+} \) phosphors was obtained at 1600°C under \( \text{N}_2 \) atmosphere. \( \text{Sr}_2\text{Si}_5\text{N}_8: \text{Eu}^{2+} \) phosphors structure were investigated using X-ray diffraction (XRD); the phosphors excitation spectrum, emission spectrum and decay lifetime were obtained by fluorescence spectrophotometer. Excitation spectrum of \( \text{Sr}_2\text{Si}_5\text{N}_8: \text{Eu}^{2+} \) phosphors showed a broad excitation bands peaking at 632 nm by both UV and blue light. There is an asymmetric wide emission peaks between 500 nm and 750 nm originated from the \( 4f^{6}5d^{1}-4f^{7} \) transitions of \( \text{Eu}^{2+} \) ion. And the lifetime of \( \text{Eu}^{2+} \) in the \( \text{Sr}_2\text{Si}_5\text{N}_8 \) host is 1919.81 ns. The CIE chromaticity coordinates of \( \text{Sr}_2\text{Si}_5\text{N}_8: \text{Eu}^{2+} \) phosphors are \((x=0.5549, y=0.4316)\).

1. Introduction
White light-emitting diodes (WLEDs) have attracted substantial attention in recent years for applications in solid-state lighting [1]. WLEDs have marked advantages over the traditional incandescent and fluorescent lamps, such as superior lifetime, higher brightness, lower energy consumption and better reliability [2]. Therefore, WLEDs have been considered to be promising candidates for the next generation of lighting sources. The WLEDs could be generated by combining the blue emission of InGaN diode chip with the yellow luminescence from \( \text{Y}_3\text{Al}_5\text{O}_{12}:\text{Ce}^{3+} \) (YAG: \( \text{Ce}^{3+} \)) phosphors [3, 4]. However, this white light has a low color rendering index (CRI) because the white light lacks sufficient red emission, which has limited the applications of WLEDs in indoor lighting and display devices. In order to obtain a high color rendering, there are alternative ways to generate warm white lights with a red component: one is to compensate the red deficiency of YAG:Ce\(^{3+}\) based WLEDs with an additional red phosphors [5], and another is to combine an ultraviolet (UV) chip with red, green, and blue-emitting (RGB) phosphors [6]. For both methods, red phosphors which can be efficiently excited by InGaN LED chip at blue and UV region are in great demand. Unfortunately, the efficiency of currently commercially used red phosphors is much lower than that of green and blue phosphors. For both approaches, in order to obtain high-quality warm white light output, more efforts should be devoted to searching for efficient red phosphors [7, 8].
In recent years, SiN₄-base nitride phosphors with red emission including nitridosilicates, nitridoaluminosilicates and other nitrides, etc., have been extensively studied as red phosphors for WLEDs, such as M₂Si₅N₈:Eu²⁺ and MA₅Si₃N₈:Eu²⁺ (M=Ca, Sr, Ba)[9–17]. Among these phosphors, Sr₂Si₅N₈:Eu²⁺ has demonstrated high luminescence intensity and very low thermal quenching. Several approaches have been tried for preparation of Sr₂Si₅N₈:Eu²⁺ phosphors, including a traditional solid-state reaction method using nitride raw materials and a reaction between metals and silicon diimide Si(NH)₂[18]. These materials are not only expensive but also very sensitive to oxygen and moisture. Sr₂Si₅N₈:Eu²⁺ phosphors can be directly prepared by a synthetic route using SrCO₃, Eu₂O₃, and Si₃N₄ as raw materials. However, the resultant product was a complex mixture of Sr₂Si₃N₈ and Sr₂SiO₄[19]. Therefore, a simple, efficient method is needed to synthesize Sr₂Si₅N₈:Eu²⁺ red phosphors with pure-phase and excellent luminescence properties.

In this work, using SrCO₃, Eu₂O₃, and Si₃N₄ as the raw material and carbon as reduction agent to synthesize Sr₂Si₅N₈:Eu²⁺ phosphors. All the raw materials used were commercially available, used low cost oxides, and the synthesis in a N₂ atmosphere. The photoluminescence properties and structure of obtained Sr₂Si₅N₈:Eu²⁺ phosphors were investigated.

2. Experimental

2.1. Sample Preparation

The Eu²⁺ doped Sr₂Si₅N₈ nitride phosphors were synthesized by carbothermal reduction and nitridation method. The raw materials using SrCO₃ (99.9%), Si₃N₄ (99.9%) and Eu₂O₃ (99.99%), and carbon (99.9%, n C/nSr=1.5) as main reduction agent, using NH₄Cl (99.5%, 4 wt %) as a flux. They were stoichiometrically weighed and mixed thoroughly in an agate mortar. Then mixtures transferred into a graphite/alumina/corundum crucible placed into a tubular furnace. The furnace chamber was pumped down to high vacuum and filled with pure N₂, and a flow rate of 1 L/min was maintained during the heating process. Hereafter the mixed raw materials were fired at 1600°C for 3 h under nitrogen atmosphere. After firing, the obtained phosphors were cooled to room temperature in a furnace under a continuous flow of N₂ gas. Then polish in an agate mortar to form the final products.

2.2. Characterization

The crystal structure of the as-synthesized powder was inspected with an X-ray powder diffraction (XRD, Rigaku Ultima IV), which enabled average depiction of the structure through pattern matching, using Cu Kα radiation (λ=0.154056 nm) at room temperature. Accelerating voltage is 40 kV, Emission current is 40 mA. The data were collected over the 2θ range of 10-80º. The step width was 0.02 º and the scanning speed was 15 º/min. The photoluminescence excitation and emission spectrum were measured at room temperature using a fluorescent spectrophotometer (Spectrofluorometer FS 5, United Kingdom) with a 500 W Xenon lamp as an excitation source. And the decay curve of the powder were measured at room temperature using a fluorescent spectrophotometer (Spectrofluorometer FS 5, United Kingdom) with a 500 W Microsecond Flash lamp as an excitation source. All spectra were collected at room temperature under same experimental conditions.

3. Results and Discussion

3.1. Structure Analysis of Sr₂Si₅N₈: Eu²⁺ Phosphors

Figure 1 shows the XRD pattern of Sr₂Si₅N₈:Eu²⁺ (x=0.02) phosphors samples prepared at 1600°C for 3 h. Through the comparison with the standard card of Sr₂Si₅N₈ (PDF card no. 85-0101), it can be seen in Figure 1 that samples are consisted by pure Sr₂Si₅N₈ phase and no apparent impurity phases were found. Illustrate that the Eu²⁺ doped Sr₂Si₅N₈ samples are indexed with an orthorhombic lattice with the space group of Pmn2₁. Moreover, the Eu²⁺ ions are effectively doped into the host lattice. It was also indicated that a little amounts Eu²⁺ ions doped did not caused any significant change in the host structure, because the close ionic radii of Sr²⁺ (0.112 nm) and Eu²⁺ (0.117 nm) and the Eu²⁺ and
Sr$^{2+}$ ion have the same valence, so the Eu$^{2+}$ enter the crystal lattice mainly account the location of Sr$^{2+}$ [20].

![Figure 1](image1.png)

**Figure 1.** The XRD pattern of Sr$_{2-x}$Si$_5$N$_8$:Eu$_x^{2+}$ (x=0.02) phosphors.

### 3.2. Photo luminescent Properties of Sr$_2$Si$_5$N$_8$: Eu$^{2+}$ Phosphors

![Figure 2](image2.png)

**Figure 2.** Excitation spectrum of Sr$_{2-x}$Si$_5$N$_8$:Eu$_x^{2+}$ (x=0.02) phosphors synthesized under 632 nm emission
Figure 3. Emission spectrum of Sr$_{2-x}$Si$_5$N$_8$:Eu$_{x}^{2+}$ (x=0.02) phosphors synthesized under 426 nm excitation.

Figure 2 shows the excitation spectrum of Sr$_{2-x}$Si$_5$N$_8$:Eu$_{x}^{2+}$ (x=0.02) phosphors synthesized under 632 nm emission. The broad excitation bands ranging from 360 nm to 500 nm were observed in the excitation spectrum and the wide absorption peaking ranging from 400 nm to 450 nm. The peaking centered at 426 nm is attributed to the direct excitation of Eu$^{2+}$ ion via the 4f$^7$-4f$^5$5d$^1$ parity allowed transition [21].

Figure 3 shows the emission spectrum of Sr$_{2-x}$Si$_5$N$_8$:Eu$_{x}^{2+}$ (x=0.02) phosphors synthesized under 426 nm excitation. It is clear that the emission spectrum exhibits an intense broad band ranging from 500 to 750 nm peaking at 632 nm, which is attributed to the transition from the 4f$^5$5d$^1$ excited states to the 4f$^7$ ground state of Eu$^{2+}$ ion.

Upon irradiation at different wavelengths within the excitation spectrum, the emission spectrum put up similar emission of Eu$^{2+}$ in peak shape and position, indicating the occurrence of energy transfer between the base material and the Eu$^{2+}$ activators. Due to the high covalent environment around Eu$^{2+}$ ion, emission of Sr$_{2}$Si$_3$N$_8$:Eu$^{2+}$ phosphor was observed at a fairly longer wavelength region. This intense broad excitation band located at the near-UV and visible light spectral range provides the basis for developing this kind of phosphors for white light-emitting diodes (WLEDs) conversion applications [22].

3.3. Luminescence Decay Lifetime of Sr$_{2}$Si$_3$N$_8$: Eu$^{2+}$ Phosphors

Figure 4. Luminescence decay curve of Sr$_{2-x}$Si$_5$N$_8$:Eu$_{x}^{2+}$ (x=0.02) phosphors ($\lambda_{ex}$=426 nm, $\lambda_{em}$=632 nm).
Figure 4 shows the luminescence decay curve of Sr$_{2-x}$Si$_5$N$_8$:Eu$_x^{2+}$ (x=0.02) phosphors monitored at $\lambda_{ex}=426$ nm, $\lambda_{em}=632$ nm. To reveal the lifetimes of as-prepared phosphors, computer-aided simulations of the experimental data were conducted, and the fitting result are also shown in Figure 4 as red curve.

The decay of the Eu$^{2+}$ is non-exponential and is considered as superposition of two components. The decay curve of Eu$^{2+}$ luminescence was fitted to two exponential decay given by the following equation:

$$I(t) = I_0 + A_1 \exp \left( -\frac{t}{\tau_1} \right) + A_2 \exp \left( -\frac{t}{\tau_2} \right)$$

(1)

Where $\tau_1$ and $\tau_2$ is the luminescence decay time, $I_0$, $A_1$ and $A_2$ is the fitting parameters. The squares sum of error of fitting $\chi^2$ is 1.79, shows to matching and achieved a good result. Finally, we can found the lifetime of Eu$^{2+}$ in the Sr$_2$Si$_5$N$_8$ host is 1919.81 ns.

3.4. The CIE chromaticity coordinates of Sr$_2$Si$_5$N$_8$: Eu$^{2+}$ Phosphors

Figure 5 shows the red emission of Sr$_2$Si$_5$N$_8$:Eu$^{2+}$ phosphors can be confirmed by the CIE (Commission International de L’ Eclairage 1931 chromaticity) coordinates from their emission spectra. Figure 5 shows upon excitation by 426 nm, the CIE chromaticity coordinates of Sr$_{2-x}$Si$_5$N$_8$:Eu$_x^{2+}$ (x=0.02) phosphors are (x=0.5549, y=0.4316).
4. Conclusion
We have reported a simple synthetic route to Sr₂Si₅N₈:Eu²⁺ based red phosphors for white light-emitting diodes (WLEDs), through the chemical reaction of SrCO₃, Si₃N₄ and Eu₂O₃ at 1600°C under N₂ atmosphere. The Eu²⁺ doped Sr₂Si₅N₈ nitride phosphors were synthesized by carbothermal reduction and nitridation method using carbon as main reduction agent, and NH₄Cl as a flux. Sr₂Si₅N₈:Eu²⁺ phosphors structure were investigated using X-ray diffraction (XRD). The XRD result indicate that the Eu²⁺ ions are effectively doped into the Sr₂Si₅N₈ host lattice. And the samples are indexed with an orthorhombic lattice with the space group of Pmn2₁. Excitation spectrum of Sr₂Si₅N₈:Eu²⁺ phosphors showed a broad excitation bands peaking at 632 nm by both UV and blue light. There is an asymmetric wide emission peaks between 500 nm and 750 nm originated from the 4f⁵5d¹-4f⁶ transitions of Eu²⁺ ion. And the lifetime of Eu²⁺ in the Sr₂Si₅N₈ host is 1919.81 ns. The CIE chromaticity coordinates of Sr₂Si₅N₈:Eu²⁺ phosphors are (x=0.5549, y=0.4316). The experimental spectroscopic results presented in this work show the Sr₂Si₅N₈:Eu²⁺ phosphors as an excellent candidate for getting red emission for White light-emitting diodes.

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