Development of *Lactobacillus kimchicus* DCY51T-mediated gold nanoparticles for delivery of ginsenoside compound K: *in vitro* photothermal effects and apoptosis detection in cancer cells

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**ABSTRACT**

We report a non-covalent loading of ginsenoside compound K (CK) onto our previously reported gold nanoparticles (DCY51T-AuCKNps) through one-pot biosynthesis using a probiotic *Lactobacillus kimchicus* DCY51T isolated from Korean kimchi. The ginsenoside-loaded gold nanoparticles were characterized by various analytical and spectroscopic techniques such as field emission transmission electron microscopy (FE-TEM), energy-dispersive X-ray (EDX) spectroscopy, elemental mapping, X-ray powder diffraction (XRD), selected area electron diffraction (SAED), Fourier-transform infrared (FTIR) spectroscopy and dynamic light scattering (DLS). In addition, DCY51T-AuNps and DCY51T-AuCKNps were resistant to aggregation caused by pH variation or a high ionic strength environment. Cell-based study confirmed that DCY51T-AuCKNps exhibited slightly higher cytotoxicity compared to ginsenoside CK treatment in A549 cells (human lung adenocarcinoma cell line) and HT29 (human colorectal adenocarcinoma cell line). Upon laser treatment, DCY51T-AuCKNps showed enhanced cell apoptosis in A549, HT29 and AGS cells (human stomach gastric adenocarcinoma cell line) compared with only DCY51T-AuNps treated cells. In conclusion, this preliminary study identified that DCY51T-AuCKNps act as a potent photothermal-therapy agents with synergistic chemotherapeutic effects for the treatment of cancer.

**Introduction**

The pharmacological effects of *Panax ginseng* Meyer (Korean ginseng), which has been regarded as one of the most acclaimed traditional Chinese herbal medicines for over 2000 years, can be largely accredited to its bioactive triterpenoid saponins [1]. These triterpenoid saponins, also known as ginsenosides, are further classified into three types: panaxadiol (PPD), panaxatriol (PPT) and oleanic acid [1]. The minor PPD-senosides, are further classified into three types: panaxadiol the major glycosylated ginsenosides [1]. The bioconjugation of ginsenosides to a specifically designed drug delivery system thus plays a significant role in remediating these limitations.

Spherical gold nanoparticles are widely reported to display biocompatibility for biological applications, especially as drug carriers. In addition to their attractive optical properties as multimodality imaging agents, colloidal gold can also provide photothermal therapy (PPT) in conjunction with laser [6, 7]. Unlike conventional physical and chemical methods that often employ hazardous reducing and stabilizing agents, green synthesis of metal nanoparticles can be exploited to eliminate toxic ligands and augment nanoparticles with bioactive natural macromolecules that may render nanoparticles biologically active for biomedical applications [8]. In view of the properties, we used DPPH-scavenging gold nanoparticles (DCY51T-AuNps) synthesized by bacteria as nanocarriers to deliver ginsenoside CK and improve its water dispersibility and anticancer effects.
The use of PTT is a recently developed technique and an advanced therapeutic strategy which utilize the near infrared laser photo absorbers to trigger heat for thermal excision of cancer cells upon NIR irradiation [9–11]. When compared with conventional therapies, the use of PTT exhibits distinct advantages in cancer therapy including enhanced specificity, reduction in invasion and precise spatial-temporal selectivity and so on [12, 13]. PTT can directly obliterate the cancer cells in early stage and local metastasis and even to treat metastasis. Several previous reports make use of PTT in treating cancer cell metastasis [14–16]. The efficacy of PTT depends on the transformation of light to heat with photothermal agents especially nanoscales agents.

In our study, DCY51T-AuNps were prepared following our previously published method via an intracellular membrane-bound mechanism by lactic acid-producing Lactobacillus kimchicus DCY51T. The utilization of fresh biomass to synthesize gold nanoparticles is preferred than intracellular enzymes due to the recovery of nicotinamide adenine dinucleotide (NADH)-dependent coenzymes involved in the reduction of Au3+ ion [17, 18]. Herein, ginsenoside CK was non-covalently bioconjugated to probiotic-mediated DCY51T-AuNps. Similarly, model drugs, such as hydrophobic doxorubicin and antibacterial rifaximin, were capable to be loaded onto DCY51T-AuNps by the same methodology. The complexity of drug by DCY51T-AuNps was further characterized by spectroscopic and analytical techniques. Drug loading was determined by liquid chromatography–mass spectrometry (LC–MS). The preliminary in vitro anticancer effects and PPT against lung (A549), colon (HT29) and stomach (AGS) adenocarcinoma cells of the biosynthesized ginsenoside CK-loaded gold nanoparticles (DCY51T-AuCKNps) were subsequently investigated. Finally, DCY51T-AuCKNps could be easily detected and developed to be a cancer cell diagnostic tool using real-time monitoring technologies.

**Experimental**

**Materials**

MRS (de Man, Rogosa and Sharpe) broth and agar were purchased from MBCell (Seoul, Republic of Korea). Analytical grade hydrogen tetrachloraurate (III) hydrate (gold salt) and hydrochloric acid (HCl), ACS reagent, (37%) were purchased from Sigma-Aldrich Chemicals (St. Louis, MO). Sodium chloride (NaCl, ≥99.50%) and sodium hydroxide (NaOH, ≥98%) were purchased from Daejung Chemicals and Metals Co., Ltd. (Shihheung-city, Korea). Ginsenoside CK (≥K,%) was acquired from Ginseng Bank, Kyung Hee University (Yongin, Republic of Korea). Doxorubicin-hydrochloride and rifaximin were purchased from Cayman Chemical Company (Ann Arbor, MI). Premade phosphate-buffered saline (20× PBS) was purchased from Biosesang (Seongnam, Republic of Korea). Methyl alcohol (MeOH, 95%) was procured from Samchun Pure Chemical Co. Ltd. (Pyeongtaek, Republic of Korea). Other chemicals were obtained from commercial suppliers and used as received.

**Cell lines**

RAW264.7 (murine leukemic macrophage cell line) and A549 cells (human lung adenocarcinoma cell line) were obtained from Korean Cell Line Bank (KCLB, Seoul, Republic of Korea). HT29 (human colorectal adenocarcinoma cell line) and AGS cells (human stomach adenocarcinoma cell line) were obtained from American Type Culture Collection (ATCC, Manassas, VA). Dulbecco’s modified Eagle medium (DMEM), foetal bovine serum (FBS), RPMI 1640 culture media and penicillin–streptomycin solution were purchased from GenDEPOT (Barker, TX). Soluble 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl-tetrazolium bromide (MTT) was purchased from Life Technologies (Eugene, OR). Other chemicals were obtained from commercial suppliers and used as received.

**Bacterial culture condition for the biological synthesis of DCY51T-AuCKNps**

The bacterial strain was isolated from traditional Korean kimchi as type strain Lactobacillus kimchicus DCY51T (KCTC 12976T, JCM 15530T) by Liang et al. [19]. The GenBank/EMBL/DDBJ accession numbers for the 16S rRNA gene and phес sequences of strain DCY51T are EU678893 and FJ904276, respectively [19]. The isolated strain was streaked on MRS agar and then incubated at 37°C for 48 h. Bacterial cells were inoculated into 100 mL MRS broth and incubated at 37°C and 100×g for 24 h. After incubation, 10 mL of the inoculated culture broth was further centrifuged at 6300×g for 5 min. Fresh biomass was collected and used for the one-pot biological synthesis of gold nanoparticles according to our previous study [20].

**Synthesis of DCY51T-AuCKNps**

DCY51T-AuNps were synthesized by L. kimchicus DCY51T according to a previous method by Markus et al. with minor modifications [20]. The collected biomass was washed with 10 mM PBS buffer and re-suspended in 25 mL of sterilized dH2O. 1 mM gold salt and 1 mg ginsenoside CK (1.6 mM, Mw 1000) were added and incubated at 37°C and shaken at 150×g in the dark. Following the addition of the gold salt solution into the microbial cell suspension in dH2O, the synthesis of DCY51T-AuCKNps was detected visually by a change in the colour of the biomass in the reaction mixtures. Thereafter, the dark purple-coloured reaction mixtures were centrifuged at 2500×g for 5 min to remove supernatant and unloaded drugs. To collect the trapped, drug-loaded nanoparticles from the bacterial cells, the bacterial pellets underwent alternating cycles of ultra-sonication and continuous centrifugation at 2500×g for 5 min. Drug-loaded DCY51T-AuCKNps were collected by high speed centrifugation at 28,000×g for 10 min and were washed exhaustively with dH2O. Finally, DCY51T-AuNps and DCY51T-AuCKNps were stored as aqueous suspensions at 4°C until needed for experiments. Solid specimens were obtained by air-drying nanoparticles overnight. In addition, commercially available drugs such as doxorubicin and rifaximin were also loaded to DCY51T-AuNps to synthesize DCY51T-DoxorubicinNps and DCY51T-RifaximinNps, respectively, using the same aforementioned protocol.
High resolution transmission electron microscopy (HR-TEM) ultra-microtome analysis of DCYS1\textsuperscript{T}-AuCKNps

Prior to HR-TEM analysis, 2.5% of glutaraldehyde in 1 M PBS buffer was added to nanoparticle suspension. The sample was washed thrice with 1 × 0.1 M PBS buffer for 5 min. The supernatant was removed and 1% OsO\textsubscript{4} was added and allowed to react for 1 h. The sample was washed thrice with 1 × 0.1 M PBS buffer for 10 min. Next, the sample was washed twice using dH\textsubscript{2}O for 5 min. Then, ethanol was added for dehydration and further steps were followed by the standard protocol [22]. Subsequently, ultra-thin-sectioned samples were prepared for HR-TEM. The sectioned samples were stained with uranyl acetate for 10 min and lead citrate for 1 min prior to analysis. The isolated strains *L. kimchicus* DCYS1\textsuperscript{T} and *L. brevis* DCY65 were used as biofactories for gold nanoparticles loaded with ginsenoside CK.

Characterization of nanoparticles

The concentration of gold nanoparticles was estimated by the equation derived by Haiss et al. whereas the particle diameter was estimated by the Debye–Scherrer formula with a shape factor (κ) of 0.9 [22, 23]. The absorbance spectra of the drug-loaded nanoparticle suspension were scanned in the range of 300–800 nm with an ultraviolet–visible (UV–vis) spectrophotometer (Ultrospec\textsuperscript{TM} 2100 pro) with a quartz cuvette having a 10 mm path length (2100 Pro, Amersham Biosciences Corp., Piscataway, NJ). Field emission transmission electron microscopy (FE-TEM) and HR-TEM were assessed by a JEM-2100F operated at 200 kV (JEOL, Peabody, MA). FE-TEM images were obtained by spreading droplets of aqueous suspension of drug-loaded DCYS1\textsuperscript{T}-AuCKNps onto a carbon-coated copper grid and drying at 60 \textdegree C. X-ray powder diffraction (XRD) spectra were obtained with a D8 Discover diffractometer with a general area detector diffraction system (Bruker, Hamburg, Germany) operating at a voltage of 40 kV and 40 mA with CuK\textalpha\textsubscript{x} radiation of 1.54 Å in the 2\theta range of 20–80\textdegree. The zeta potential and hydrodynamic particle size of the drug-loaded DCYS1\textsuperscript{T}-AuCKNps suspended in dH\textsubscript{2}O were determined with an ELS-Z2 series zeta potential and particle size analyser (Otsuka Electronics Co. Ltd., Osaka, Japan) at 25°C. Pure water with a refractive index of 1.3328, viscosity of 0.8878 and dielectric constant of 78.3 was used as referred.

Photoluminescence spectra of DCYS1\textsuperscript{T}-AuCKNps

The photoluminescence (PL) spectra were measured at room temperature using a fixed 325 nm line of a HeCd laser as the excitation source. Emitted light was collected by a lens and analysed using a grating monochromator and a GaAs photomultiplier tube. Standard lock-in detection techniques were used to maximize the signal-to-noise ratio. The laser power for the PL excitation was 5.66 W/cm\textsuperscript{2}.

Drug loading efficiency

The drug loading of DCYS1\textsuperscript{T}-AuCKNps was determined by LC–MS. The amount of CK loaded onto the nanoparticles was quantified by LC–MS. Briefly, 3 mg of air-dried drug-loaded DCYS1\textsuperscript{T}-AuNps was suspended in dimethyl sulphoxide (DMSO)–MeOH mixture (1:4, v/v). The nanoparticle suspension was sonicated overnight to ensure complete release of drug into MeOH mixture. The supernatant was collected by filtration and the amount of released drug was quantified using an LC–MS instrument (Agilent Technologies 6410 Triple Quad, Kromasil C18 column (2.1 mm × 100 mm, 2.7 \textmu m), Santa Clara, CA) with distilled water (solvent A) and MeOH (solvent B) as mobile phases at 5% B and 95% A for 0 min, 60% B and 40% for 15 min, 100% B for 18 min, 100% B for 30 min, 5% B and 95% A for 32 min and 5% B and 95% A for 32 min, all at a flow rate of 0.4 mL/min. The injection volume of each sample was 1 \textmu L. UV detection was performed at 203 nm. Drug loading efficiency was calculated by the following formula:

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\text{Drug loading efficiency (\%)} = \frac{\text{Total amount of drug released from nanoparticles}}{\text{Yield of nanoparticles}} \times 100\%
\]

Estimation of gold ion content in DCYS1\textsuperscript{T}-AuCKNps by ICP/QMS

The gold content of DCYS1\textsuperscript{T}-AuCKNps was measured by an inductively coupled plasma/quadrupole-mass spectrometer (ICP/QMS) using a previously published method [27]. Analysis was carried out using a NexION 300D (PerkinElmer, Waltham, MA). Briefly, 0.5 g of DCYS1\textsuperscript{T}-AuCKNps were hydrolysed with

ence. Fourier-transform infrared (FTIR) spectra of ginsenoside CK and the corresponding drug-loaded DCYS1\textsuperscript{T}-AuCKNps were recorded on a PerkinElmer Spectrum 100 spectrometer (PerkinElmer, Waltham, MA). The air-dried nanoparticle powder was scanned in the range of 4000–450 cm\textsuperscript{−1} at a resolution of 4 cm\textsuperscript{−1} on KBr pellets. Furthermore, the in vitro stability of DCYS1\textsuperscript{T}-AuNps and DCYS1\textsuperscript{T}-AuCKNps over a wide range of pH and electrolyte conditions was assessed. The stability of nanoparticles under different pH and ionic strength conditions is an important criterion for their biomedical application [25]. The intracellular pH of tumour tissues (pH 5.0–6.5) is often more acidic than the physiological pH conditions of normal tissues (pH 7.4) [26]. The stability of DCYS1\textsuperscript{T}-AuCKNps in an aqueous suspension was demonstrated over a wide range of pH conditions (pH 2–10) by adjustment with 0.1 M HCl and 0.2 M NaOH at room temperature. In addition, DCYS1\textsuperscript{T}-AuCKNps was also suspended in different electrolyte conditions (10\textsuperscript{−1} to 10\textsuperscript{−5} M) by addition of NaCl. The lack of changes in the surface plasmon resonance (SPR) peak and band, as measured by UV–vis spectrophotometer after incubation, indicates the colloidal stability of nanoparticles in all of the above conditions [8].
HCl and heated at 180°C for drying. Then, a mixture of HCl and HNO₃ solution was added to the suspension to reach a final volume of 10 mL. A radio frequency (RF) power of 1600 W, a coolant gas flow rate of 18 L/min, an auxiliary gas flow rate of 1.30 L/min and a nebulizer gas flow rate of 1.02 mL/min were used for the analysis.

In vitro cell cytotoxicity of DCY51T-AuCKNps

Cytotoxicity analysis was conducted via MTT assay in RAW264.7, HaCaT, A549 and HT29 cells according to our previous studies [28, 29]. To investigate the cytotoxicity of drug-loaded DCY51T-AuNps and free ginsenoside CK, the cells were treated with various concentrations of drug-loaded DCY51T-AuCKNps and free drug at 37°C for 48 h at 90% confluence. Finally, the absorbance of each coloured solution was quantified by an enzyme-linked immunosorbent assay reader (Bio-Tek Instruments, Inc., Winooski, VT) at 570 nm. The optical density of formazan formed in untreated cells (negative control) represented 100% cell viability.

Furthermore, the antiproliferative activity of DCY51T-AuCKNps towards the human stomach cancer cell lines (AGS) was evaluated using an MTT assay after treatment with an IR beam. After adhesion, DCY51T-AuCKNps at different concentrations were added to 96 well plates. Based on the preliminary test results, the anti-proliferative activity of DCY51T-AuCKNps was determined with three different concentrations ranging from 25 to 100 μL in 0.1% DMSO. Before treatment, all wells except the control well were illuminated with a laser beam for approximately 3 min at 800 nm. The culture plates were incubated for 48 h in a 37°C humidified atmosphere containing 5% CO₂. The plates were then washed with 100 μL PBS and 100 μL of 0.05% MTT reagent was added to each well and then incubated for 4 h at the same conditions as stated previously. The MTT solution was removed and 200 μL DMSO was added to each well. Finally, the plate was shaken in a microplate shaker for 10 min to dissolve the purple formazan crystals. 5-Flourouracil served as a positive control and was similarly formulated. The DMSO solution was used as a negative control.

In vitro photothermal therapy and Hoechst nuclear staining assay

RAW264.7, AGS, A549 and HT29 cells were seeded onto glass coverslips and incubated in six-well plates at a density of 2.5 × 10⁵ cells/mL for 24 h at 37°C in a humidified incubator with 5% CO₂ and 95% air. The cells were then incubated with a suspension of DCY51T-AuNps or DCY51T-AuCKNps (1 or 5 mg/mL; equivalent to 0.18 and 0.89 μM of ginsenoside CK) for 24 h. Next, the cells were washed with 1× PBS to remove excess nanoparticles and fixed with 3.7% (v/v) formaldehyde for 5 min at room temperature. The cells were further incubated in fresh culture medium for 24 h. Subsequently, RAW264.7, A649 and HT29 cells were irradiated with a 635 nm laser (Shanghai Dream Laser Technology Co. Ltd., Shanghai, China) with a heat flux density of 0.74 W/cm² directly above the cell plates for 10 min with a laser spot size of 0.2 cm², resulting in a total energy dose of 88.80 J. On the other hand, AGS cells were irradiated with an 800 nm laser. After irradiation, the cells were maintained at 37°C in a humidified incubator with 5% CO₂. Finally, the irradiated cells were stained by Hoechst 33258 solution (2 μg/mL) for 20 min in the dark at room temperature to analyse synergistic cell apoptosis [30]. Images of Hoechst stain were acquired by a fluorescence microscope (×400, Optinity, Korean Labtech, Namyangju, South Korea). In all studies, cells without nanoparticles and cells incubated with DCY51T-AuNps at 5 μg/mL served as blank and negative control and were subjected to
the same irradiation conditions as cells incubated with DCY51\textsuperscript{-}AuCKNps.

**Total internal reflection scattering (TIRS)/differential interference contrast (DIC) detection system**

TIRS microscopy was carried out on an upright Olympus BX51 microscope (Olympus Optical Co., Ltd., Tokyo, Japan) equipped with a 100\times oil iris objective lens (NA = 0.6–1.3, UPLANFLN, Olympus Optical Co., Ltd., Tokyo, Japan). TIRS illumination was used, with a 30 mW, 637 nm laser (MGL-III-637–200 mW, Changchun New Industries Optoelectronics Tech. Co., Ltd., Changchun, China) for illumination of the gold nanoparticles. A Uniblitz mechanical shutter (model LS3220-R3, Vincent Associates, Rochester, NY) and a driver (model VMM-D1, Vincent Associates, Rochester, NY) were synchronized to the electron-multiplying cooled charge-coupled device camera (512\times512 pixel imaging array, QuantEM 512SC, Tucson, AZ). The exposure time was 100 ms and wavelength selection was accomplished with a central wavelength of 620/14 nm purchased from Semrock (Rochester, NY). A lab-made DIC system was modified from previously published configurations [31]. The illumination light was provided by a 100 W halogen lamp. All images were obtained with the MetaMorph 7.5 software (Universal Imaging, Sunnyvale, CA).

**Statistical analysis**

All experiments were performed at least in triplicates unless stated otherwise. Values shown are reported as means ± SDs. The statistical significances between control and sample groups were evaluated by Student’s t-test. A greater extent of statistical significances was assigned with increasing number of asterisks (*p* < 0.05, **p** < 0.01, ***p*** < 0.001 and ****p*** < 0.0001).

**Results and discussion**

**One-pot biosynthesis of DCY51\textsuperscript{T}-AuCKNps**

Following the addition of 1 mM of gold salt and 1 mg ginsenoside CK into the biomass suspension, the colour of the cell biomass changed to pinkish purple at 4 h and then consequently became deep purple after 12 h. No further colour change was observed after 12 h of incubation. This observation was consistent with our previous study [20]. The growth medium alone and cell biomass without gold salt (negative control) did not exhibit any colour changes. The colloidal nanoparticles excite the free electrons and may result in oscillations, which reverberate with the frequency of visible light wavelengths. This SPR phenomenon is responsible for the absorption in the visible spectrum and thus the distinct purple colour of the synthesized DCY51\textsuperscript{T}-AuNps and DCY51\textsuperscript{T}-AuCKNps [32].

Upon further investigation by HR-TEM (Figure 1(A)), the formation of DCY51\textsuperscript{T}-AuCKNps was more concentrated upon the bacterial capsule and cell wall than on the cytoplasmic membrane, possibly due to the reduction of the metal ions by the enzymes and exopolysaccharides (EPSs) present in the cell wall and slime layer. Few of the nanoparticles were found to be transported into the cytoplasmic membrane where they further aggregated. If the reaction could proceed for a longer time, more diversity in the size of...
DCY51\(^T\)-AuCKNps was found \([17, 31, 32]\). Our sample was collected after 12 h of incubation. As anticipated, negative control did not exhibit the formation of gold nanoparticles. For comparison, \(L. \text{ brevis} \) DCY65 was also used as biofactories for gold nanoparticles synthesis (Figure 1(B)). Similarly, to \(L. \text{ kimchicus} \) DCY51\(^T\), the formation of DCY65-AuCKNps was localized on the surface of bacterial cells with some aggregated in the cell wall.

Up to now the exact mechanism of green synthesis has not yet been fully elucidated, even though some reports have presented the intracellular synthesis of AuNps by \(Lactobacillus\) species \([17, 19, 31]\). It has been postulated that EPSs and NADH-dependent oxido-reductase secreted by the microorganism on the bacterial membranes were responsible for the reduction of gold ions, and membranous phospholipid, organic acids and amino acid residues inside the cells were responsible for providing stabilization of the nanoparticle's carboxylate groups via electrostatic interaction. The metal nuclei and bacterial membranous substances interacted to form nanoparticles through further reduction and aggregation \([18, 32]\). Several non-covalent interactions, including electrostatic interactions, hydrogen bonds, hydrophobic forces and van der Waals forces, were conjectured to play key roles in the binding process of proteins with ginsenoside CK to the surface of DCY51\(^T\)-AuNps \([25]\). Furthermore, poorly water soluble ginsenoside CK might also move into the hydrophobic micro-regions of the proteins stabilizing the nanoparticles \([34]\). In addition to DCY51\(^T\)-AuCKNps, DCY51\(^T\)-DoxorubicinNps (Figure 1(C)) and DCY51\(^T\)-RifaximinNps (Figure 1(D)) were also synthetized by \(L. \text{ kimchicus} \) DCY51\(^T\). Similarly, gold nanoparticles were found inside the cell envelope or on the surface. Some were formed in the loose slime layers that had trailed off into the medium. This observation suggested that metal ion reduction and stabilization might be localized within cell envelope or cell surface by enzymes or EPS present in the cell wall and slime layer.

**FE-TEM analysis of DCY51\(^T\)-AuCKNps, DCY51\(^T\)-DoxorubicinNps and DCY51\(^T\)-RifaximinNps**

DCY51\(^T\)-AuCKNps (Figure 2(A)), DCY51\(^T\)-DoxorubicinNps (Figure 2(B)) and DCY51\(^T\)-RifaximinNps (Figure 2(C)) were further analysed by FE-TEM. The FE-TEM images showed spherical DCY51\(^T\)-AuCKNps, DCY51\(^T\)-DoxorubicinNps and DCY51\(^T\)-RifaximinNps with varying sizes of 10–40 nm, 20–60 nm and 10–20 nm, respectively. Figure 2(a,f,k) reveals that the DCY51\(^T\)-AuCKNps and DCY51\(^T\)-RifaximinNps were spherical while the DCY51\(^T\)-DoxorubicinNps aggregated further to form flower-shaped nanostructures. The distributions of gold were clearly visible in the electron image of DCY51\(^T\)-AuCKNps (Figure 2(c)), DCY51\(^T\)-DoxorubicinNps (Figure 2(h)) and DCY51\(^T\)-RifaximinNps (Figure 2(m)), respectively. Energy-dispersive X-ray (EDX) spectroscopy of DCY51\(^T\)-AuCKNps (Figure 2(e)), DCY51\(^T\)-DoxorubicinNps (Figure 2(j)) and DCY51\(^T\)-RifaximinNps (Figure 2(o)) determined the highest
optical absorbance band peaks to be at 2.3 keV, which corresponds to the characteristic peak of metallic gold [35].

**Particle size distribution of DCY51T-AuCKNps, DCY51T-DoxorubicinNps and DCY51T-RifaximinNps**

The size distribution profile of DCY51T-AuCKNps (Figure 3(A)), DCY51T-DoxorubicinNps (Figure 3(B)) and DCY51T-RifaximinNps (Figure 3(C)) was determined using a particle size analyser with respect to intensity, number and volume. Similar to our previously reported DCY51T-AuNps, DCY51T-AuCKNps, DCY51T-AuDoxorubicinNps and DCY51T-AuRifaximinNps synthesized by strain DCY51T were not entirely monodisperse as observed from the FE-TEM results in Figure 2 [20]. This polydispersity was also apparent from the wide size distribution of the intensity-based histogram (Figure 3(A–C)). The average hydrodynamic diameter of DCY51T-AuNps was determined to be 134.25 ± 1.63 nm with a PDI of 0.13 ± 0.001. Contrary to our previously published dynamic light scattering (DLS) result of DCY51T-AuNps [20], the Z-average value of the drug-loaded nanoparticles demonstrated a modest growth. This behaviour was also observed in DCY51T-AuDoxorubicinNps (Figure 3(B)) and DCY51T-AuRifaximinNps (Figure 3(C)). This result indicated the loading of ginsenoside CK, doxorubicin and rifaximin on the surface of DCY51T-AuNps [36]. At neutral pH in dH2O, the zeta potential (ζ) of DCY51T-AuNps and DCY51T-AuCKNps was recorded to be ~51.04 mV and ~52.22 mV, respectively, implying that the nanoformulations were stable in the aqueous suspension. The small increase in the positive charge indicated that other attractive forces, such as van der Waals forces or hydrogen bonding, might be responsible for facilitating the drug loading process [27]. According to previous studies, nanoparticles of ≤200 nm may accumulate in tumour tissues due to the enlarged gap junctions (0.1–2 μm) in the poorly aligned and defective endothelia lining. This enhanced permeability and retention (EPR) effect is the main mechanism of non-targeted drug delivery system [37]. Furthermore, the cationic charge of DCY51T-AuCKNps may induce higher cellular uptake and greater cytotoxicity to cancer cells due to the permeability of the anionic cell membrane [38].

**UV–vis spectra analysis, XRD profile and photoluminescence spectroscopy of DCY51T-AuCKNps**

UV–vis spectroscopy was used to monitor the visible absorption bands of the nanoparticles. These absorption bands are strongly dependent on the nanoparticle size, shape, surface and agglomeration state [22, 39, 40]. Nanoparticles with larger crystallite sizes or particle agglomeration often have broad absorption bands spanning into the red-shifted range (>600 nm) due to the presence of both transversal and longitudinal SPRs [41]. Likewise, the extinction spectra of DCY51T-AuDoxorubicinNps exhibited a major absorbance peak at ~540 nm after 12 h of incubation at 37°C (Figure 4(A)). There was no change in the SPR peak between drug-loaded nanoparticles (DCY51T-AuCKNps) and DCY51T-AuNps, meaning that there was no significant change in the size, shape, surface and agglomeration state of the two nanoparticles. The sharpness of the SPR peaks of DCY51T-AuNps and DCY51T-AuCKNps revealed that the biosynthesized nanoparticles might be suitable for bio-imaging tags in dark-field microscopy techniques [30]. For comparison, we display the absorption of whole
cells and a sample where nanoparticles were continuously reduced within or on the surface of bacterial cells [33]. As expected, whole cells of *L. kimchicus* DCY51T did not contribute to the peaks observed in DCY51T-AuNps and DCY51T-AuCKNps. In addition, the lack of change in the SPR peak and band, as measured by UV–vis spectrometry, after incubation at different pH values and ion strengths indicated the good colloidal stability of the nanoparticles in all of the conditions (Figure 5(A–D)) [30]. Furthermore, DCY51T-AuNps and DCY51T-AuCKNps were stable for 3 months at room temperature with no significant changes in absorbance values or aggregation states in the suspensions (Figure 5(E,F)). These results confirmed the colloidal stability of DCY51T-AuNps and DCY51T-AuCKNps by *L. kimchicus* DCY51T in all of the above experimental conditions. To boot, the colloidal stability of DCY51T-AuNps and DCY51T-AuCKNps in aqueous medium was substantiated by the relatively high ζ-potential; high magnitude of ζ-potential indicates that the nanoparticles will resist aggregation resulted by electrostatic repulsion between nanoscale particles [42].

The crystalline nature of the biologically synthesized nanoparticles was evaluated by XRD analysis. Figure 4(B) demonstrates the intensive diffraction patterns of biosynthesized DCY51T-AuCKNps. The four characteristic peaks of gold were indexed to the (111), (200), (220) and (311) lattice planes of Bragg’s reflection. The most intense peak at (111) proves that DCY51T-AuCKNps were composed of (111) lattice plane. The selected area electron diffraction (SAED) pattern likewise confirmed the polycrystalline orientation of the nanoparticles as shown in the multi-faceted rings in Figure 2(D). The average metallic diameter of DCY51T-AuCKNps was estimated by taking the full width at half maximum (FWHM) of the most intense peak (111); the resulting nanoparticles maintained an average crystallite diameter of ~6.04 nm. Photoluminescence spectroscopy was performed. The intensity of DCY51T-AuCKNps at 425 nm is shown in Figure 4(C).

**FTIR profile analysis of DCY51T-AuCKNps**

In our previous study by Markus et al., we proposed that EPS and/or enzymes secreted on the bacterial membrane and amino acid residues inside the cells were responsible for playing both the roles of reducing agent and stabilizers to DCY51T-AuNps via electrostatic interactions between the oxidized carboxylate (COO−) groups of the enzymes and the surface of nanoparticles (Au-COO−) [20]. Additionally, several absorption peaks of *L. kimchicus* DCY51T and DCY51T-AuNps were also present in the FTIR spectrum of DCY51T-AuCKNps (tabulated in Figure 6), namely the strong characteristic peak of carbonyl (C=O) stretching of amide at 1650 cm−1 and the medium characteristic peak of amine (N–H) bending at

![Figure 5. UV–vis absorbance spectra of the obtained nanoparticles depicting their high stability in various conditions due to protein capping. Stability of DCY51T-AuNps and DCY51T-AuCKNps against varied pH conditions (A, B) and electrolytic concentrations (C, D) and three-month storage (E, F), respectively.](image-url)
1540 cm⁻¹ contributed by the proteins and amino acid residues of L. kimchicus DCYS1° [43–45].

The complexation of ginsenoside CK on the surface of DCYS1°-AuCKNps was confirmed by the peaks corresponding to alkane (C–H) bending at 1500–1440 cm⁻¹ and aliphatic ether (C–O) stretching at 1300–1000 cm⁻¹ of ginsenoside CK [46]. These peaks were suspected to be the result of the non-covalent attractive forces (i.e. electrostatic interactions, hydrogen bonds, hydrophobic forces and van der Waals forces) between ginsenoside CK and DCYS1°-AuNps: following...
nucleation and stabilization by surface-bound enzymes, ginsenoside CK might form stable interactions with the amino acid residues on the surface of DCY51T-AuNps. In addition, the hydroxyl groups of ginsenoside CK might act as additional reducing species for the reduction of Au$^{3+}$ ions into Au$^{0}$ and oxidize into carboxylates as shown in the inset of Figure 6.

**Drug loading efficiency via LC–MS and Au ion content**

Quantification of the concentration of ginsenoside CK in the drug-loaded DCY51$^{T}$-AuCKNps is a critical parameter to evaluate the properties of the nanoformulations (Figure 7). DCY51$^{T}$-AuCKNps were exposed to a DMSO-MeOH mixture (1:4, v/v) and sonication at room temperature overnight to allow complete release of ginsenoside CK and evaluation of the drug loading efficiency. The majority of the released drug was expected to be dissolved in the MeOH fraction because of the high solubility in MeOH [47]. Following sonication, approximately 110.30 μg of ginsenoside CK per one mg of DCY51$^{T}$-AuCKNps was released in the medium. A calibration curve was prepared of a CK standard with known concentrations in the range of 10–100 μg/mL (regression equation: $y = 1,203,623.2951x + 8,433,298.2623$; $R^2 = 99.86\%$). Thus, the drug loading efficiency was determined to be $\sim 11.03\%$. The released CK from DCY51$^{T}$-AuCKNps was subjected to mass spectroscopy to determine its molecular weight. The molecular weight of the ginsenoside obtained from DCY51$^{T}$-AuCKNps was determined to be 618.3 which is similar to the molecular weight of CK (622.88) (mass spectrum of CK, $m/z = 664.3 = [\text{MW} + \text{formic acid}], \text{MW} = 618.3$). The Au (Mw = 197) content of DCY51$^{T}$-AuCKNps was 104 μg/g DCY51$^{T}$-AuCKNps, as measured by ICP/OES [26].

**In vitro cell cytotoxicity of DCY51$^{T}$-AuCKNps by MTT assay**

The results of MTT assay for DCY51$^{T}$-AuNps are shown in Figure 8. We observed a different *in vitro* cytotoxicity profiles in RAW264.7 and HT29 than those previously observed in our prior study [20]. We conjectured that washing the biosynthesized gold nanoparticles with 80% MeOH, which was conducted in our prior study, might promote the dynamic unbinding of bioactive compounds from the nanoparticles surface and thus reducing their biological activity. Additionally, DCY51$^{T}$-AuNps were air-dried and re-dispersed in water before analysis; air-drying might cause the nanoparticles to irreversibly aggregate which influenced their toxicities. In view of these factors, DCY51$^{T}$-AuNps and DCY51$^{T}$-AuCKNps were washed with dH$_2$O and stored as aqueous suspensions at 4°C to prevent unwanted decomplexation and aggregation prior to analyses.

Upon close observation, the viability of the cells in RAW264.7 and HaCaT cell lines bounced back as the viability of the cells decreased significantly at 1 μg/mL (69.38 ± 1.80% and 82.32 ± 1.47%, respectively) and started to increase to 93.33 ± 1.94% and 81.42 ± 7.51%, respectively, at 100 μg/mL (Figure 8(A,B)). Moreover, cell viability of A549 cells was significantly inhibited to 68.18 ± 1.05% at 1 μg/mL, 64.40 ± 1.27% at 10 μg/mL, 68.18 ± 2.44% at 50 μg/mL and
finally to 74.24 ± 2.32% at 100 μg/mL (Figure 8(C)). On the other hand, HT29 cells showed a dose-dependent cell inhibition as the cell viability was inhibited to 81.90 ± 2.96% at 1 μg/mL and finally to 57.67 ± 5.30% at 100 μg/mL (Figure 8(D)).

The cytotoxicity of DCY51T-AuCKNPs and standard ginsenoside CK was evaluated in RAW264.7 and cancer (A549 and HT29) cell lines via MTT assay. The results are shown in Figure 8. Sufficient concentrations of DCY51T-AuCKNPs were used such that the concentration of CK in the free drug was equivalent to that in the drug-loaded nanoparticles. Based on MTT results, DCY51T-AuCKNPs were found to exert less cytotoxicity compared to free CK in RAW264.7 (Figure 8(A)). At 20 μM of equivalent CK content (or 113 μg/mL of DCY51T-AuCKNPs), DCY51T-AuCKNPs inhibited the cell viability to 70.74 ± 2.49% while free drug inhibited the cell viability to 51.60 ± 7.37%, suggesting that the systemic cytotoxicity of CK in RAW264.7 cells was reduced in DCY51T-AuCKNPs (Figure 8(A)). However, DCY51T-AuCKNPs demonstrated greater cell inhibition in RAW264.7 cells starting at 10 μM (75.24 ± 5.41%) of equivalent ginsenoside CK content compared to A549 (85.87 ± 4.32%) and HT29 (90.91 ± 1.70%) cells. Moreover, loading of ginsenoside CK seemed to reduce the overall inhibition activity of DCY51T-AuNPs in HT29 cells. DCY51T-AuCKNPs exhibited slightly higher cytotoxicity in A549 and HT29 cells compared to free drug starting at 5 μM of equivalent CK content, which did not exert significant cytotoxicity in RAW264.7 cells (Figure 8(B)). The antiproliferative activity of DCY51T-AuCKNPs towards the stomach cancer cell line (AGS) was also evaluated using an MTT assay after treatment with an IR beam (Figure 9(A)). Purple formazan crystals formed in control wells and IR-irradiated wells (negative control wells), whereas the DCY51T-AuCKNPs-treated wells showed little purple colour (50 μL), slight discolouration (100 μL) and finally colourless solution (250 μL) after IR treatment suggesting that cell viability was inhibited in cancer cells in a dose-dependent manner (Figure 9(B)). Additionally, disintegrated cancer cells after treatment of DCY51T-AuCKNPs and IR beam in a dose-dependent manner was also observed (Figure 9(B)).

For the following in vitro PPT experiment, the use of DCY51T-AuCKNPs at 1 and 5 μg/mL (equivalent to 0.18 and 0.89 μM of CK) was selected since within this concentration range DCY51T-AuCKNPs did not exert strong cytotoxicity towards RAW264.7 cells as observed in Figure 9(C–E).

**In vitro photothermal therapy and apoptosis detection**

As the result of the EPR effect, DCY51T-AuCKNPs may accumulate in tumour tissues due to leaky vasculatures. Once accumulated, cationic DCY51T-AuCKNPs may be taken up inside the cells by receptor-independent endocytosis through interaction with the cell membrane [48]. Ginsenoside CK may then be released by lysosomal degrading enzymes, while DCY51T-AuNPs are expected to retain their stability in a low acidic environment. The detection of intracellular DCY51T-AuCKNPs in stomach cancer cells (AGS) presumably uptake via receptor-independent endocytosis can be observed in the images in Figure 10. Differential interference contrast and dark field images showed clearly intracellular DCY51T-AuCKNPs particles from treated AGS cell lines (Figure 10(C)), whereas untreated AGS cell lines never showed any traces of intracellular DCY51T-AuCKNPs. In addition, DCY51T-AuCKNPs can be irradiated at mild wavelength to rapidly induce cell lysis due to the elevated temperature induced by light-to-heat conversion (hyperthermia). Localized heat treatment utilizing temperatures over 40 °C has been reported to damage tumour cells irreversibly by inducing protein degradation and
Figure 10. Detection of intercellular DCYS1\(^T\)-AuCKNps in the stomach cancer cell lines (AGS). (A) Control sample, (B) 800 nm laser treatment for 10 min. (C) DCYS1\(^T\)-AuCKNps + laser at 800 nm for 10 min.

Figure 11. Fluorescent images of cells stained with Hoechst stain after an incubation time of Ankrum h. (A) Blank. (B) DCYS1\(^T\)-AuNps; 5 \(\mu\)g/mL. (C) DCYS1\(^T\)-AuCKNps; 1 \(\mu\)g/mL. (D) DCYS1\(^T\)-AuCKNps; 5 \(\mu\)g/mL. Apoptotic cells are indicated with white arrows. The scale bar equals 10 \(\mu\)m. There were more apoptotic cells in the DCYS1\(^T\)-AuCKNps groups than in blank or in the negative control (DCYS1\(^T\)-AuNps). The co-treatment groups of hyperthermia and chemotherapy exhibited the greatest number of apoptosis in cancer cells.
DNA repair deficiency [49]. To investigate the photothermal effects of DCY51T-AuCKNps, a 635 nm laser at a heat flux density of 0.74 W/cm² was used to irradiate RAW264.7 cell and cancer cell (A549 and HT29) lines. AGS cells were irradiated with 800 nm laser. To specifically analyse the photothermal effects, Hoechst staining was applied. As shown in Figure 11, the fluorescent images showed that the blank group, which was irradiated with the laser treatment alone, exhibited no apoptosis. As predicted, incubation with DCY51T-AuNps (negative control) induced a slight apoptosis in the RAW264.7 and cancer cells (A549 and HT29) because they caused cell inhibition within this concentration range (5 μg/mL), as demonstrated by MTT results.

However, the number of apoptotic cells in the DCY51T-AuCKNps groups was significantly higher than those in the DCY51T-AuNps groups at the same concentration (5 μg/mL). Moreover, cells incubated with DCY51T-AuCKNps (both 1 and 5 μg/mL) were shown to demonstrate higher apoptosis in stomach, lung and colon adenocarcinoma cells than RAW264.7 cells. Increasing the concentration of DCY51T-AuCKNps from 1 to 5 μg/mL also resulted in higher apoptotic cells in cancer cell lines. We have also established increasing cell apoptotic mediated cell death in stomach cancer cell line (AGS) through a dose-dependent manner by staining the AGS cell lines with propidium iodide (Supplementary Figure S1). The cells treated with DCY51T-AuCKNps and laser treatment caused cell damages which led to dead cells. Staining the dead cells with propidium iodide resulted in the appearance of red fluorescence.

This finding suggests that DCY51T-AuCKNps have the potential to increase apoptosis in cancer cells in conjunction with photo-induced hyperthermia more than chemotherapy alone, possibly due to the increased sensitivity of tumour cells to ginsenoside or light-triggered drug release into cancer cells [6, 7, 49]. The proposed mechanism of cell internalization of DCY51T-AuCKNps by the EPR effect and in vitro PPT is demonstrated in Figure 12.

Several studies have reported that when conjugated with drugs, gold nanoparticles can function as drug carriers to enhance the release and uptake of drugs by tumour tissues undergoing hyperthermia stimulation [50]. Despite the lower toxicity of gold nanoparticles compared with other metallic nanoparticles, their lack of clearance from the body still remains an issue [50]. Specific targeting of cancer cells and minimizing the risk of in vivo toxicity are the main challenges of this study and must be thoroughly addressed prior to in vivo experiments [51].

Conclusions

In summary, this study highlights the development of ginsenoside CK-gold nanocarriers (DCY51T-AuCKNps) via an intracellular membrane-bound mechanism by lactic acid-producing Lactobacillus kimchicus DCY51T. Ginsenoside CK
was effectively loaded onto the surface of gold nanoparticles by non-covalent conjugation such as, electrostatic forces, hydrogen bonding, hydrophobic forces and van der Waals interactions. Upon further investigation by HR-TEM, the particles were formed in the bacterial capsule and cell wall than on the cytoplasmic membrane, possibly due to reduction of the metal ions by enzymes and EPSs present in the cell wall and slime layer. The drug loading efficiency was determined to be ~11.03%. The FTIR spectra of DCY51\textsuperscript{T}-AuCKNPs demonstrated the presence of alkane (C–H) bending and ether (C–O) stretching of ginsenoside CK, further confirming the direct complexation of ginsenoside to the surface of nanoparticles. The in vitro stability of DCY51\textsuperscript{T}-AuNps and DCY51\textsuperscript{T}-AuCKNps demonstrated their resistance to aggregation and dissociation caused by pH variation or a high ionic strength environment. Although DCY51\textsuperscript{T}-AuCKNPs demonstrated greater cell inhibition in RAW264.7 cells at 10–20 μM of equivalent ginsenoside CK concentration compared to A549 and HT29 cells at the same concentrations, preferential cytotoxicity of DCY51\textsuperscript{T}-AuCKNPs against A549 cells and HT29 compared to free ginsenoside CK was demonstrated by MTT assays. Increased apoaptosis of DCY51\textsuperscript{T}-AuCKNPs in cancer cells compared to cell apoptosis count in RAW264.7 cells and DCY51\textsuperscript{T}-AuNps-treated cells at sublethyal concentrations was confirmed by Hoechst staining with excitations by laser. These results suggest that DCY51\textsuperscript{T}-AuCKNps are promising drug delivery platforms for cancer therapy as novel photothermal and chemotherapeutic agents.

**Disclosure statement**

No potential conflict of interest was reported by the author(s).

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