Ozone in the Upper Stratosphere and Mesosphere

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A detailed photochemical model of the upper stratosphere and mesosphere is compared with three extensive sets of ozone observations: Atmospheric Explorer-E backscattered ultraviolet experiment (BUV), Nimbus-4 BUV, and rocket flights from Wallops Flight Center (ROCOZ). The Nimbus-4 and rocket observations are most sensitive to ozone between 30 and 50 km, whereas observations from AE-E measure the abundance of ozone up to 70 km. The photochemical model accurately reproduces the observed relationship between BUV intensity and local solar zenith angle, although the absolute calibration on AE-E appears to be in error. The AE-E observations and the model both exhibit a morning-afternoon asymmetry, with more ozone in the morning owing to the build up of HO\textsubscript{3} species in the afternoon. Seasonal changes in atmospheric temperature produce an annual maximum in tropical mesospheric ozone during June-July-August. The amplitude of the observed effect is somewhat larger than calculated by the model. Some problems appear to remain with the presently accepted kinetic rates for HO\textsubscript{3} species in the atmosphere.

1. INTRODUCTION

Absorption of ultraviolet sunlight by ozone and molecular oxygen provides the energy which maintains the temperature structure and wind field of the middle atmosphere [Murgatroyd and Singleton, 1961; Leovy, 1964]. The daily heating is maximal near the stratopause and induces diurnal and semi-diurnal tides which are important to the structure and vertical transports of the upper mesosphere and thermosphere [Lindzen, 1979; Forbes and Garrett, 1979]. Thus the distribution of ozone plays a major role in determining the mean circulation of the upper atmosphere.

The concentration of ozone in the upper stratosphere and lower mesosphere is controlled by photochemical production and loss. Radicals containing hydrogen (OH, HO\textsubscript{3}, H), chlorine (Cl, CIO), and nitrogen (NO, NO\textsubscript{2}, NO\textsubscript{3}) all play important roles through reactions with each other, ozone and with atomic oxygen [Chapman, 1930; Bates and Nicolet, 1950; Hampson, 1964; Wofsy and McElroy, 1974]. As discussed in section 2, these reactions proceed so rapidly that transport of ozone and other radicals plays a minor role in determining the ozone distribution between 35 and 80 km.

The present research focuses on chemical models and observational data in the upper stratosphere and the mesosphere. This region of the atmosphere contains less than 25% of the total ozone but is often assumed to be the photochemical source region for ozone in the lower atmosphere [Mahlman et al., 1980]. The complex and poorly characterized molecules which are of interest in the lower stratosphere (e.g., HOCl, CINO\textsubscript{3}, HOONO\textsubscript{2}, N\textsubscript{2}O\textsubscript{5} [National Academy of Sciences, 1979]) are not important in the photochemistry above 35 km, where three-body reactions proceed slowly and the ultraviolet radiation field is relatively more intense.

This paper examines three recently available sets of observations of ozone in the stratosphere and mesosphere. Each data set is comprised of many individual observations acquired over periods longer than a year. Two of these data sets (Atmospheric Explorer-E and Nimbus-4) are based on satellite monitoring of backscattered ultraviolet sunlight (BUV). The third set consists of rocket measurements of ozone from the differential absorption of sunlight as a function of altitude (rocket ozone project [ROCOZ] [Krueger and Wright, 1979]).

A primary question which must be answered is whether these three data sets provide a consistent picture of ozone in the upper stratosphere and mesosphere.

The analysis of the complete set of data from Atmospheric Explorer-E (BUV) and a limited set of Nimbus-4 data [Heath et al., 1979] is presented in section 3, where differences between the two data sets are discussed with attention to seasonal variations. Results from the theoretical model are compared in turn with the ozone concentrations of the mid-latitude ROCOZ data and with the satellite albedos from the tropics.

A major goal of this research is the development of an objective photochemical model which, as accurately as possible, reproduces all aspects of the observed distribution of ozone. In section 4 the overall validity of this photochemical model is critically examined. The key uncertainties in the model are pointed out, and weaknesses or inconsistencies among the observational data for ozone are discussed.

2. PHOTOCHEMICAL MODEL

The photochemistry which controls the abundance of ozone in the upper stratosphere and mesosphere can be described by about 50 kinetic and photolytic reactions as given in Table 1. The definitions of odd-oxygen (O\textsubscript{3} = O\textsubscript{3} + O + NO\textsubscript{2} + CIO) and odd-hydrogen (HO\textsubscript{3} = H + OH + HO\textsubscript{3}) encompass those species which rapidly interchange with one another. These are displayed schematically in Figure 1. The photochemical lifetime for each family ranges from hours to days and is shown in Figure 2.

a. Chemistry

The rates of chemical reactions which produce or destroy odd-oxygen are given in Figure 3. The only significant source of odd-oxygen above 30 km is the photolysis of molecular oxygen [Chapman, 1930].

(R38) \( \text{O}_2 + \text{hv} \rightarrow \text{O} + \text{O} \)

The dominant sinks for O\textsubscript{3} in the mesosphere (45–80 km) are the reactions of atomic oxygen with members of the HO\textsubscript{3} family [Bates and Nicolet, 1950].

(R4) \( \text{O} + \text{OH} \rightarrow \text{H} + \text{O}_2 \)

(R5) \( \text{O} + \text{HO}_3 \rightarrow \text{OH} + \text{O}_2 \)
In the upper stratosphere (30–45 km), the major losses of odd-oxygen are due to NO, catalytic cycles [Hanson, 1964],

\[(R23) \quad NO + O_3 \rightarrow NO_2 + O_2\]

and to Cl, catalytic cycles [Stolarski and Cicerone, 1974; Wofsy and McElroy, 1974],

\[(R33) \quad Cl + O_3 \rightarrow ClO + O_2\]

\[(R35) \quad ClO + O \rightarrow Cl + O_2\]

and to the self-reaction [Chapman, 1930],

\[(R2) \quad O + O_3 \rightarrow O_2 + O_2\]

The reaction rate constant \(k\) for all of the above reactions are taken from the recent NASA-JPL [1979] review. The largest uncertainties (±50%) are associated with \(k_a\) and \(k_s\). The photolysis rate for molecular oxygen is calculated from the solar flux and cross-section data in Table 2 and from the treatment of the Schumann-Runge bands described in Logan et al. [1978]. The resulting \(O_2\) dissociation rates are similar (±10%) to those of Frederick and Hudson [1980].

The odd-oxygen budget of the mesosphere is strongly

\[
\begin{array}{c}
\text{ODD OXYGEN} \\
\text{ODD HYDROGEN}
\end{array}
\]

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Table 1. Photochemical Reactions

| Reactions | Rate (cm\(^3\) s\(^{-1}\) or cm\(^6\) s\(^{-1}\)) | Note* |
|-----------|---------------------------------|-------|
| \((R1)\)  | \(O + O_2 + M = O_3 + M\) | 1.05E-34 exp (+5100/T) LPWM |
| \((R2)\)  | \(O + O_3 + O_2\) | 1.5E-11 exp (+2218/T) NASA |
| \((R3)\)  | \(O + O + M \rightarrow O_2 + M\) | 9.6E-34 exp (+480/T) LPWM |
| \((R4)\)  | \(O + HO_2 \rightarrow OH + O_2\) | 4.0E-11 NASA |
| \((R5)\)  | \(O + H_2 \rightarrow OH + H\) | 3.5E-11 NASA |
| \((R6)\)  | \(O_3(D) + N_2 \rightarrow O + N_2\) | 5.3E-11 exp (-5100/T) LPWM |
| \((R7)\)  | \(O_3(D) + O = O + O_2\) | 2.0E-11 exp (+107/T) NASA |
| \((R8)\)  | \(O_3(D) + O = O + O_2\) | 2.9E-11 exp (+67/T) NASA |
| \((R9)\)  | \(O_3(D) + H_2O = OH + OH\) | 2.3E-10 NASA |
| \((R10)\) | \(O_3(D) + H_2 = OH + H\) | 0.99E-10 NASA |
| \((R11)\) | \(O_3 + H = OH + O_2\) | 1.4E-10 exp (-470/T) NASA |
| \((R12)\) | \(O_3 + OH = H + O_2\) | 1.6E-12 exp (-940/T) NASA |
| \((R13)\) | \(O_3 + HO_2 = OH + O_2\) | 9.6E-13 NASA |
| \((R14)\) | \(O_3 + H_2 = OH + H\) | 1.90E-32 exp (+350/T) LPWM |
| \((R15)\) | \(O_3 + OH = HO_2 + O_2\) | 4.2E-10 exp (-350/T) LPWM |
| \((R16)\) | \(O_3 + OH = OH + O_2\) | 4.2E-10 exp (-350/T) LPWM |
| \((R17)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R18)\) | \(O_3 + OH = OH + O_2\) | 4.0E-11 NASA |
| \((R19)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R20)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R21)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R22)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R23)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R24)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R25)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R26)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R27)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R28)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R29)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R30)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R31)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R32)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R33)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R34)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R35)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R36)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |
| \((R37)\) | \(O_3 + HO_2 = OH + O_2\) | 4.0E-11 NASA |

*NASA refers to NASA-JPL [1979]. LPWM refers to Logan et al. [1978].

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*Fig. 1. Photochemical reactions for the mesosphere and the upper stratosphere.*
coupled to the chemistry of odd-hydrogen [Bates and Nicolet, 1950; Hunt, 1966; Liu and Donahue, 1974]. Figure 4 presents the rates of specific reactions which control concentrations of HOx species. Production of odd-hydrogen is dominated by

\[(R9) \quad \text{O}^{('D')} + \text{H}_2\text{O} \rightarrow \text{OH} + \text{OH}\]

below 60 km and by

\[(R42) \quad \text{H}_2\text{O} + \text{hv} \rightarrow \text{H} + \text{OH}\]

above 65 km. Below 70 km, over 90% of the HOx loss is due to

\[(R17) \quad \text{OH} + \text{HO}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2\]

Above 70 km, significant HO loss occurs through

\[(R15) \quad \text{H} + \text{HO}_2 \rightarrow \text{H}_2 + \text{O}_2\]

The rates for the HO production reactions are reasonably well determined [Baulch et al., 1980; NASA-JPL, 1979], but the reaction rates \(k_{17}\) and \(k_{15}\) remain uncertain. The NASA-JPL [1979] value for \((R17)\), \(k_{17} = 4.0 \times 10^{-11} \text{ cm}^3 \text{ s}^{-1}\), is based on low pressure measurements [Burrows et al., 1977; Chang and Kaufman, 1978] and is much smaller than the values at atmospheric pressure reported by DeMore [1979], \(k_{17} \approx 1 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}\). The measurements made at a pressure of 2–4 torr employed inert carrier gases with at most traces of molecular oxygen. If we accept DeMore’s hypothesis of a pressure-dependent reaction rate, then the present kinetic data do not determine rates applicable in the atmosphere.

Because \((R17)\) is the dominant path for odd-hydrogen recombination below 75 km, a change in \(k_{17}\) has the same effect on model calculations as a change of opposite sign in \(\text{H}_2\text{O}\), the primary source for odd-hydrogen. In particular, the standard kinetic model with \(k_{17} = 4 \times 10^{-11} \text{ cm}^3 \text{ s}^{-1}\) and 2 ppm \(\text{H}_2\text{O}\) would produce the same HOx distributions below 75 km as a model with DeMore’s rate \(k_{17} \approx 1 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}\) and 5 ppm \(\text{H}_2\text{O}\).

The rate constant for \((R15)\) has been measured only at room temperature [Hack et al., 1978, 1979], and thus the temperature dependence adopted here [Baulch et al., 1972] is not well determined. This reaction \((R15)\) is most important to the HOx budget near the extremely cold mesopause \((-190^\circ\text{K})\). If this reaction has no temperature dependence, then it proceeds twice as fast at the mesopause as compared to the expression in Table 1. The concentrations of HOx are reduced accordingly, and the \(\text{O}_2\) abundance should increase near the mesopause.

b. Time Dependent Calculations

The diurnal variations of the concentrations of chemical species \(n_i\) (cm\(^{-3}\)) over the 24-hour day are calculated from the continuity equation,

\[
\frac{dn_i}{dt} = P_i - L_i + \nabla \cdot \Phi_i \tag{1}
\]

where \(P\), \(L\), and \(\nabla \cdot \Phi\) are the local production, local loss, and divergence of the flux for species \(i\), respectively (cm\(^{-3}\) s\(^{-1}\)). For short-lived species \(P\) and \(L\) will be much larger than \(\nabla \cdot \Phi\). Consequently, the calculations presented here assume that transport of radicals is negligible, and the flux divergence term in (1) is neglected for all species.

The local production and loss terms in (1) involve products of concentrations of other chemical species. The coupled set of continuity equations is accordingly nonlinear. The inverse Euler technique is used to solve (implicitly) for species concentrations at each new time step. The mass of any closed system is conserved to 1 part in \(10^6\). Time steps of the order of 1 hour are used to model the 24-hour day with finer temporal resolution near sunrise and sunset. Steady state conditions are achieved by an acceleration procedure which ensures that the net change in concentration of all species is zero after a 24-hour model day [Wofsy, 1978].

c. Solar Radiation and BUV Albedos

The radiation field is calculated at each time step during the model day for the true zenith angle of the sun. The model includes absorption by molecular oxygen and ozone as well as molecular scattering. The attenuation of the direct solar flux is calculated by integrating through the atmosphere in the direction of the sun. The calculation takes into account both sphericity and changes in local solar time along the path. Refraction is not included. This direct solar flux is the source function used to calculate the diffuse radiation in an inhomogeneous scattering atmosphere which is assumed to be plane parallel [Prather, 1974]. The method allows the calcula-
TABLE 2. Solar Flux and Cross Sections

| λ (Å)  | Flux*          | O₂*          | O₃          | H₂O          | ϝ†       |
|--------|----------------|--------------|-------------|--------------|----------|
| 1215.6 | 2.51 x 10⁻¹¹  | 1.0 - 20     | 1.43 - 17   |              |          |
| 1700   | 2.01 x 10⁻¹²  | 8.8 - 19     | 4.2 - 18    |              |          |
| 1750   | 2.94 x 10⁻¹²  | 2.7 - 19     | 2.7 - 18    |              |          |
| 1800   | 5.32 x 10⁻¹²  | SR bands     | 0.96 - 18   | 7.5 - 19     |          |
| 1850   | 6.90 x 10⁻¹²  | SR bands     | 0.69 - 18   | 5.0 - 20     |          |
| 1900   | 9.09 x 10⁻¹²  | SR bands     | 0.93 - 18   | 5.2 - 21     |          |
| 1950   | 2.40 x 10⁻¹²  | SR bands     | 0.41 - 18   | 6.3 - 22     |          |
| 2000   | 3.75 x 10⁻¹²  | SR bands     | 0.31 - 18   | 6.3 - 23     |          |
| 2050   | 5.71 x 10⁻¹²  | 1.0 - 23     | 0.34 - 18   |              |          |
| 2100   | 1.48 x 10⁻¹³  | 8.5 - 24     | 0.53 - 18   |              |          |
| 2150   | 2.10 x 10⁻¹³  | 6.5 - 24     | 1.02 - 18   |              |          |
| 2200   | 2.59 x 10⁻¹³  | 5.0 - 24     | 1.82 - 18   |              |          |
| 2250   | 3.21 x 10⁻¹³  | 4.0 - 24     | 2.94 - 18   |              |          |
| 2300   | 3.56 x 10⁻¹³  | 3.0 - 24     | 4.51 - 18   |              |          |
| 2350   | 3.66 x 10⁻¹³  | 1.6 - 24     | 6.34 - 18   |              |          |
| 2400   | 3.76 x 10⁻¹³  | 1.0 - 24     | 8.26 - 18   |              |          |
| 2450   | 3.86 x 10⁻¹³  | 7.0 - 25     | 1.00 - 17   |              |          |
| 2500   | 3.97 x 10⁻¹³  | 3.0 - 25     | 1.11 - 17   |              |          |
| 2550   | 5.18 x 10⁻¹³  | 1.2 - 25     | 1.17 - 17   |              |          |
| 2600   | 8.54 x 10⁻¹³  |              | 1.11 - 17   |              |          |
| 2650   | 1.41 x 14     |              | 9.86 - 18   |              |          |
| 2700   | 1.59 x 14     |              | 7.85 - 18   |              |          |
| 2750   | 1.44 x 14     |              | 5.83 - 18   |              |          |
| 2800   | 1.74 x 14     |              | 3.99 - 18   |              |          |
| 2850   | 2.33 x 14     |              | 2.47 - 18   |              |          |
| 2900   | 3.15 x 14     |              | 1.37 - 18   |              |          |
| 2950   | 3.89 x 14     | 0.76 - 18    |              | 1.00         |          |
| 3000   | 3.85 x 14     | 0.38 - 18    |              | 1.00         |          |
| 3050   | 4.28 x 14     | 0.18 - 18    |              | 0.95         |          |
| 3100   | 7.84 x 14     | 0.99 - 19    |              | 0.45         |          |
| 3200   | 1.32 x 15     | 0.24 - 19    |              | 0.00         |          |
| 3300   | 1.59 x 15     | 0.69 - 20    |              | 0.00         |          |
| 3400   | 1.69 x 15     | 1.71 - 21    |              | 0.00         |          |
| 3500   | 1.80 x 15     | 0.43 - 22    |              | 0.00         |          |
| 3600   | 2.99 x 15     |              |              | 0.00         |          |
| 3800   | 4.19 x 15     |              |              | 0.00         |          |
| 4000   | 2.27 x 16     |              |              | 0.00         |          |
| 5000   | 4.85 x 16     |              |              | 0.00         |          |
| 6000   | 7.38 x 16     |              |              | 4.60 - 21    | 0.00     |

In squared centimeters.

Sources: Flux, Vernazza et al. [1976], Hinteregger [1979]; O₂, see Logan et al. [1978]; O₃, Griggs [1968], quantum yield of O(1D) from NASA-JPL [1979]; H₂O, Watanabe et al. [1953], Thompson et al. [1963]; NO₂, NASA-JPL [1979]; CH₄, HCl, ClO, HO₂, see Logan et al. [1978].

* Solar flux in units: photons cm⁻² s⁻¹ interval⁻¹ at 1 AU. Cross sections in units: cm².
† Read 2.51 x 10⁻¹¹ as 2.51 x 10⁻¹¹.

The calculations assume that the only source of scattered light is due to Rayleigh and Raman scattering by the molecules of the background atmosphere (N₂, O₂, CO₂). The mean Rayleigh scattering cross section per molecule is given by [Penndorf, 1957]

$$\sigma_{n, \lambda} = \frac{8}{3} \pi \lambda^{-4} (n_s - 1)^2 N_s^{-2}$$

where \(\lambda\) is the wavelength (Å), \(N_s\) is the molecular density at 15°C (cm⁻³), and \(n_s\) is the index of refraction for air at 15°C [Edlen, 1953].
Rotational Raman scattering will contribute significantly to the measured quantity of scattered light if the incident source is nearly continuous with wavelength or if the source is monochromatic and the instrument cannot resolve the rotational Raman shift of up to 20 Å. The Raman scattered light consists of an unshifted component as well as two wings which are distributed according to the population of rotational states of the scattering gas. For air between 200°K and 300°K the relative intensities of the Stokes ($\Delta J = +2$), central ($\Delta J = 0$), and anti-Stokes ($\Delta J = -2$) components are 0.43, 0.26, and 0.31, respectively. Penney et al. [1974] attribute the entire depolarized component of Rayleigh scattered light to rotational Raman scattering. The total cross section for this process is

$$\sigma_{RRS} = 1.39 \times 10^{-29} \lambda^{-4} \text{cm}^2$$  (6)

where the additional factor of $-0.5$ in the exponent of $\lambda(\mu m)$ is indicated by the wavelength dependence of the anisotropy [Penney et al., 1974]. The cross section for vibrational Raman scattering [Hyatt et al., 1973; Penney et al., 1974] is approximately an order of magnitude smaller than that for rotational Raman and may be ignored in the present context.

For a given wavelength in the interval 2500–3000 Å, rotational Raman scattering contributes an additional 5.9% to the Rayleigh extinction cross section but only an extra 1.5% to the scattered light at the incident wavelength, $\lambda_0$. The remaining 4.4% is scattered at a shifted wavelength into wings of width 10 Å on either side of $\lambda_0$. If we ignore this shift in wavelength, the cross section and phase function for Rayleigh + Raman scattering by air can be defined as

$$\sigma_{m} = 5.40 \times 10^{-21} \lambda^{-4} (\mu m)^2$$  (7)

$$P_m(x) = 0.7625 + 0.7124 \cos^2 x$$  (8)

where $x$ is the scattering angle. These expressions are similar to Penndorf’s [1957] formulation with a depolarization of $\rho = 0.035$. The error induced in the 2500–3000 Å albedos by neglecting the finite bandwidth of the scattered light should not exceed 5% and will in general be much less. Cross sections for ozone absorption and molecular scattering at the wavelengths used in backscattered ultraviolet observations are given in Table 4.

d. Model Results and Discussion

The relative importance of chemistry versus transport of ozone can be estimated by comparing the photochemical lifetime of odd oxygen with the effective time scales for transport. As shown in Figures 2 and 5, odd-oxygen in the stratosphere is dominated by ozone; its lifetime is approximately 10 days at 30 km and decreases to less than 1 day at 40 km. On the basis of the time scales for vertical diffusion of trace gases [cf., Huntien, 1975; Logan et al., 1978; Wofsy, 1978] and the maximum inferred meridional winds [Leovy, 1964; Mahlman and Moxim, 1978], we do not expect transport of odd-oxygen to play a dominant role in determining the concentrations of O$_3$ above 35 km. Local photochemical production and loss may even control the ozone density below 35 km when vertical transport is inefficient, for example during summer at mid-latitudes [Mahlman et al., 1980]. The family of odd-hydrogen radicals has a much shorter stratospheric lifetime and will be in photochemical steady state.

Odd oxygen in the upper mesosphere is comprised mostly of atomic oxygen, and its lifetime increases above 75 km, reaching approximately 1 day at 90 km. Important transport of odd oxygen through both tidal displacement vertical mixing [Keneshea et al., 1979] may be expected at these altitudes. The amplitude of the atmospheric tides grows exponentially with altitude, and vertical mixing is rapid, corresponding to an eddy diffusion coefficient of the order of $10^6 \text{cm}^2 \text{s}^{-1}$ [Hunten, 1975; Keneshea et al., 1979; Allen et al., 1980]. Significant transients are observed in the abundances of atomic oxygen in these regions [Wasser and Donahue, 1979]. Odd-hydrogen lifetimes also increase rapidly with altitude above 75 km, and we may no longer presume the HO$_2$ family to be in photochemical steady state above about 80 km.

The concentration of H$_2$O in the upper stratosphere and mesosphere is very important to the chemical model, but data above 35 km are sparse, and we must rely on stratospheric measurements to place limits on mesospheric H$_2$O. Our choice for the H$_2$O profile is constrained by the range of observations as reviewed by Harries [1976] and by recent measurements in the tropical stratosphere [Kuhn, 1979; Kley et al., 1979]. A uniform mixing ratio of 5 ppmv H$_2$O is adopted from 30 to 90 km, but values from 2 to 10 ppmv are also considered.

The choice of a uniform mixing ratio for H$_2$O is based on the assumption that there are no significant sources or sinks for H$_2$O in the region 35–75 km. In the lower stratosphere the H$_2$O mixing ratio is expected to increase by 3 ppmv between the tropopause and the stratopause owing to oxidation of CH$_4$. Above 75 km, H$_2$O is destroyed by conversion to H$_2$ through a sequence of reactions beginning with

$$\text{H}_2\text{O} + hp \rightarrow \text{H} + \text{OH}$$  (R42)

and culminating with

$$\text{H} + \text{HO}_2 \rightarrow \text{H}_2 + \text{O}_2$$  (R15)

The resulting molecular hydrogen is photochemically inert in the mesosphere [Liu and Donahue, 1974] except below 60 km where O(1D) concentrations are sufficient to convert H$_2$ into odd-hydrogen ((R10)) and subsequently water ((R17)).

Profiles of the trace gases CO, CH$_4$, H$_2$, and the chemical families Cl and NO$_x$ were derived from Logan et al. [1978] and are given in Table 3. Unless otherwise stated, all calculations are for equinoctial conditions with the 15°N atmosphere from the U.S. Standard Atmosphere Supplements [1966].

The calculated noontime concentrations of O$_3$ and O are shown in Figure 5. The sensitivity of the odd-oxygen chemis-
try to the concentration of H_2O increases with altitude. Indeed, a factor of 5 decrease in H_2O above 70 km causes up to an order of magnitude increase in O_3 and O_2.

The detailed diurnal behavior of ozone with altitude is displayed in Figure 6. Significant day-to-night variations in the concentration of ozone occur above 50 km because the daytime atomic oxygen density is comparable to or greater than that of ozone. The daily cycle of ozone above 70 km exhibits a complex structure with variation of the ozone concentrations in excess of a factor of 2 over the sunlit portion of the day.

The rise in ozone and total odd oxygen begins approximately 20 minutes after sunrise, driven by photolysis of molecular oxygen ((R38)) in the Schumann-Runge bands. Photolysis of H_2O at Lyman-alpha, the source of odd hydrogen above 65 km, is strongly peaked about local noon. The HOx concentrations maximize in the afternoon, and the ozone density is thus asymmetrical about noon. An increase in the concentration of mesospheric water depresses the mean concentration of ozone and causes the peak daytime concentration to occur earlier in the day. In this photochemical regime, odd oxygen will also be sensitive to changes in solar Lyman-alpha. Such variations are known to occur [Hinteregger, 1979] over the solar rotation (~30 days) and over the solar cycle (~11 years; see section 3g). Noontime densities of OH and H for the standard model are shown in Figure 7. Also shown is the decrease in OH for 2 ppm H_2O and the increase in OH corresponding to solar maximum.

3. OZONE OBSERVATIONS

a. BUV Satellites

Remote sensing of the earth's atmosphere by satellite provides the only practical means for global monitoring of the constituents of the stratosphere and mesosphere. The ultraviolet spectrometers on the Atmospheric Explorer-E and Nimbus-4 satellites have measured the backscattered solar flux intermittently since 1970. These observations contain a wealth of information about the ozone mixing ratios in the upper atmosphere and provide a primary test of the photochemical model.

The AE-E satellite was launched on November 20, 1975, into an equatorial orbit with an inclination of approximately 20°. The orbit of AE-E has been changed from an initial perigee-apogee range of 150–3000 km to a nearly circular orbit at 250 km and eventually raised to a 450 km circular orbit. For an observing pass over the sunlit hemisphere, the backscattered ultraviolet (BUV) instrument is pointed toward the nadir so that it observes the solar flux reflected from the atmosphere in the direction of the zenith. Thus, during a suitable period, the AE-E satellite can observe the BUV radiances from 20°S to 20°N latitude over all longitudes and for all local solar times from sunrise to sunset.

The Nimbus-4 satellite was launched into an 1100 km circular orbit and began BUV operations in April 1970. The satellite orbit is nearly synchronous with local solar time, monitoring local noon conditions on the daylit hemisphere and local midnight on the dark side. Thus, the solar zenith angles represented in the Nimbus-4 observations are restricted to noontime values, and their variation is directly correlated with the seasonally varying declination of the sun.

The BUV spectrometer on both AE-E and Nimbus-4 [Heath et al., 1973] detects the diffuse radiances (erg cm^{-2} s^{-1} Å^{-1} sr^{-1}) with a field of view 12° in diameter and measures the solar irradiance (erg cm^{-2} s^{-1} Å^{-1}) by observing the sun through a transmission filter (AE-E) or by measuring light from a diffuse reflector (Nimbus-4). The ratio of backscattered light to solar flux defines a dimensionless BUV albedo (3). The instrument operates in 12 discrete wavelength channels, of which only the five shortest will be considered here: 2555, 2735, 2830, 2876, and 2922 Å. At these wavelengths the BUV albedos are not affected by clouds or by ground albedo. The solar fluxes measured by the AE-E BUV instrument and the relevant molecular cross sections are given in Table 4.

![Fig. 5. Vertical profiles of noontime ozone and atomic oxygen.](image-url)

For the standard photochemical model with 15°N annual atmospheric, the horizontal bars show the effect of water vapor (2-5-10 ppm) on the concentrations.

![Fig. 6. Diurnal variation of ozone with altitude. The ozone concentrations have been calculated with the standard model under equatorial and equinoxal conditions. The assumption of a local steady state may not be valid below 36 km or above 80 km.](image-url)
The shorter wavelengths have greater ozone extinction cross sections, and consequently the solar flux is absorbed at high altitudes. The peak contribution to $p_\lambda$ for the range of wavelengths shown in Figure 8 varies from 40 to 66 km with wavelength and local solar zenith angle. The broad maxima of the sensitivity function $dp_\lambda/dz$ make it difficult or impossible to determine uniquely the ozone concentration at a specific altitude from a set of BUV albedos. Thus the inversion of BUV data requires a model profile for ozone which introduces some a priori assumptions into the retrieved profiles [cf., DeLuisi et al., 1979].

All the BUV data have been analyzed here in terms of the albedo as a function of the solar zenith angle. In the tropical and subtropical mesosphere, the variation in ozone with season is not great since both the ozone source and sink terms are affected similarly by solar elevation. For the same reason the diurnal variation in ozone tends to follow the solar zenith angle rather than the local time.

b. Atmospheric Explorer-E Observations

The Atmospheric Explorer BUV observations began in December 1975 and continued through March 1977, when the BUV instrument failed. The instrument became operational again in December 1978, and observations began anew. The calibrated radiances were calculated from the raw telemetry data with a program supplied by B. Guenther and R. Dasgupta (private communication, 1978). All the AE-E orbits through 1977 which are catalogued to contain BUV short wavelength data have been examined. Those observations which meet the following criteria were selected for study: (1) Observations must be made below 800-1000 km to avoid contamination by particle radiation [Heath et al., 1973]; (2) between 15% and 80% of the solar flux is absorbed at the top of the atmosphere near sunrise/sunset; and (3) for local solar zenith angles greater than 80°, observations must be made below 800 km because of the large gradient of BUV with longitude. For the two extreme wavelengths shown in Figure 8 varies from 40 to 66 km with wavelength and local solar zenith angle. The broad maxima of the sensitivity function $dp_\lambda/dz$ make it difficult or impossible to determine uniquely the ozone concentration at a specific altitude from a set of BUV albedos. Thus the inversion of BUV data requires a model profile for ozone which introduces some a priori assumptions into the retrieved profiles [cf., DeLuisi et al., 1979].

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The set of AE-E BUV data which fulfills the above requirements consists of 256 orbits with over 11,000 individual obser-

![Fig. 7. Vertical profiles of noon-time OH and H. The standard model with 5 ppm H$_2$O is given by the solid line; the model with 2 ppm H$_2$O, the dashed line; and the model at solar maximum with 5 ppm H$_2$O, the asterisks.](image)

**TABLE 4. BUV Wavelength Data**

| $\lambda$, Å | $F_{\lambda}$ (Sun at 1 AU), erg cm$^{-2}$ s$^{-1}$ Å$^{-1}$ | $\alpha$(O$_3$)$_{\lambda}$, $\times 10^{-19}$ cm$^2$ | $\alpha$(air)$_{\lambda}$, $\times 10^{-26}$ cm$^2$ |
|-------------|-----------------------------------------------|-------------------------------|-------------------|
| 2555        | 10.8                                          | 11.7                          | 11.4              |
| 2735        | 22.8                                          | 6.38                          | 8.45              |
| 2830        | 35.8                                          | 2.96                          | 7.29              |
| 2876        | 36.3                                          | 1.84                          | 6.80              |
| 2922        | 57.1                                          | 1.02                          | 6.35              |

Sources: Second column, B. Guenther (private communication, 1980); third column, Inn and Tanaka [1953], DeMore and Raper [1964], Griggs [1968]; fourth column, Edlen [1953], Penndorf [1957], Penney et al. [1974].

These solar fluxes are solely for use within the AE-E BUV dataset and are applied here only to determine the BUV albedo.

vations at each wavelength. Figure 9 presents the statistical frequency of BUV observations for the wavelength 2555 Å; other wavelengths show similar but not identical distributions. Although the month-by-month distribution of BUV observations is irregular—with certain months not represented at all—we see no significant selection effect in latitude versus season. Likewise, the distribution of observations with local solar time is fairly uniform if we average over a 3-month observing period.

The monthly mean albedos observed by Atmospheric Explorer-E are shown in Figures 10–14 for all five wavelengths. The statistical analysis of the AE-E data was performed by fitting the observed albedos to a piecewise linear function of solar zenith angle. The BUV values reported here refer to the midpoints of specific intervals in solar zenith angle.

The calculation of albedo from the spacecraft-observed specific intensities requires knowledge of the solar flux at the time of observation. The eccentricity of the earth’s orbit (0.0167) causes a 6.6% variation in the solar flux at the top of the atmo-

![Fig. 8. Contribution to the BUV albedo ($\rho$) as a function of altitude. For the fixed ozone profile shown in Figure 5, the absolute contribution to the albedo at the five standard BUV wavelengths is shown for solar zenith angles of 30° and 80°. For the two extreme wavelengths 2555 and 2922 Å, the dominant source region of the backscattered intensity is seen to rise with increasing solar zenith angle.](image)
sphere with a maximum at perihelion in January. In the initial analysis the BUV albedos were computed with a constant mean solar flux as measured by the instrument and given in Table 4. This procedure induces a similarly phased oscillation of 6.6% in the albedos reported in Figures 10–14. In all subsequent comparisons, the variation in solar flux has been included in both the photochemical model and the observed albedos.

The Nimbus-4 BUV albedos from the entire first year of observations were made available by A. Fleig and V. G. Kaveeswar (private communication, 1979). The calculation of the monthly mean albedos as a function of solar zenith angle was performed in the manner described for the AE-E data. Two selection processes are specific to the Nimbus-4 data: (1) only observations between 25°S and 25°N latitude are included, and (2) all data from a substantial region centered over the South Atlantic Ocean have been discarded. Particle radiation above 1000 km causes an excess signal [cf., Heath et al., 1973] which is readily discerned on a latitude-longitude grid and which extends westward from 60°E to 150°W longitude. The tropical Nimbus-4 data are limited to solar zenith angles less than 48°. Thus albedos are reported only for solar zenith angles of 10° and 30°, corresponding to intervals (0°,
Each monthly mean albedo shown in Figures 10-14 is based on 800 to 3000 observations.

d. Direct Comparison of BUV Albedos

Two striking features of Figures 10-14 are the systematic difference between the AE-E and Nimbus-4 albedos and the large, apparently annual cycle in the BUV radiances. The Nimbus-4 albedos are consistently 40-60% larger than the AE-E albedos. In spite of efforts to discern the cause of this difference (B. W. Guenther, private communication, 1980) none has been found. As expected, the annual cycle shows the 6.6% solar flux variation but, in many cases, has a larger amplitude. It is seen in both sets of BUV albedos and seems to be associated with the mesosphere above 50 km (i.e., short wavelengths and large solar zenith angles are most strongly affected).

Previous studies of the Atmospheric Explorer BUV [Frederick et al., 1977, 1978; Frederick, 1979] considered only small samples of the entire data set. Also, these studies did not discuss the problems associated with the absolute calibration of the AE-E albedos. Published studies of the Nimbus-4 BUV [e.g., DeLuisi et al., 1979] have presented only the inverted ozone profiles and have not analyzed the BUV radiances.

e. Rocket Ozone Observations

Additional ozone data obtained by rocket soundings are available for comparison with model calculations. The rocket ozone project (ROCOZ) launched rockets from Wallops Island, Virginia (38øN) at the rate of one per month during the period March 1976 to March 1979 [Wright et al., 1978; Krueger and Wright, 1979]. The optical ozonesonde used in these flights measured the attenuation of solar flux at four wavelength channels between 2500 and 3200 Å as the instrument descended on a parachute from its release at 58-70 km. The observed extinction of sunlight provides a measure of the ozone number density above the altitude of the ozonesonde.

The average profile for ozone from 32 to 58 km as reported by Wright [Krueger and Wright, 1979] is shown in Figure 15 along with two theoretical profiles of noontime ozone. The modeled distributions represent the mean of profiles computed for the January, April, July, and October 30øN atmospheres [Cole and Kantor, 1978]. The standard photochemical model diverges from the observed profile above 40 km. However, the model with only 2 ppm H2O agrees well with the ROCOZ data, predicting uniformly 8% less ozone over the range of 5 scale heights in ozone density.

The annual variation of ozone at Wallops Island is displayed in Figure 16. Concentrations of ozone observed for each season are satisfactorily reproduced by the model above 40 km where the photochemical lifetime of ozone is less than 1 day. The calculated seasonal differences at these altitudes are due mainly to the structure of the background atmosphere. For example, if ozone mixing ratios were to be plotted against pressure, winter O3 would exceed summer O3 between 0.8 and 3 mbar (~40 to 50 km) with the reverse true above 0.8 mbar. Below 40 km the model fails to predict the observed spring maximum and fall minimum. This discrepancy may be due either to the poorly characterized mean reference atmosphere or to the large rates of ozone transport, as seen in the models of Mahlman et al. [1980].

f. Absolute Calibration of AE-E Albedos

Figure 17 shows a comparison between the BUV albedos from AE-E and Nimbus-4 with those calculated from the ROCOZ data. The figure includes also albedos derived from a daytime ozone profile by Hilsenrath [1971] and from high-altitude twilight ozone profiles measured by solar occultation with the AE-E BUV instrument [Guenther et al., 1977]. The low ultraviolet albedos observed by AE-E disagree with all the other data shown in Figure 17 and cannot be explained in terms of any reasonable theoretical model. While only the occultation studies truly overlap with the AE-E BUV data, the midlatitude ROCOZ observations were made during the same period, and the Nimbus-4 albedos cover the same tropical regions at an earlier time. This evidence strongly suggests the possibility of a systematic error in the AE-E ratio of backscattered radiance to solar irradiance. The ratios of AE-E to Nimbus-4 BUV are (0.62, 0.70, 0.73, 0.74, and 0.75) for the wavelengths (2555, 2830, 2876, and 2922 Å) at a solar zenith angle of 30°. A reduction of this magnitude in BUV albedo corresponds to a factor of 2 enhancement in ozone concentrations.

g. Response to Solar Variations

The response of the atmosphere to changes in the incident solar flux is complex and may alter the distribution of ozone. In addition to the annual cycle in total flux, the sun is observed to vary over time scales corresponding to solar rotation.
Fig. 15. Mid-latitude ozone profiles. Two modeled ozone profiles are compared with the 3-year mean ozone concentrations from the ROCOZ project [Krueger and Wright, 1979] and with the Krueger and Minzner [1976] mid-latitude model. The model values represent a four season mean of noontime ozone concentrations computed by using the standard photochemical model in Table 1 for 30°N atmospheres with both 2 and 5 ppm volume mixing ratio of H2O.

(30 days) and to the solar cycle (11 years). Flux variations during the latter two phenomena depend strongly on wavelength as shown in Figure 18. The present analysis considers only the direct effect of flux variations on the photochemistry and assumes an unchanging atmospheric structure.

The 11-year solar cycle may contribute to the systematically different BUV albedos observed by AE-E (solar minimum) and Nimbus-4 (solar maximum). Data on the variation of solar flux previous to cycle 21 are sparse and are sometimes contradictory [Simon, 1978]. Such data from solar cycle 20 have been used in previous studies [Penner and Chang, 1978; Callis et al., 1979] to postulate significant variability in the solar flux shortward of 3000 Å. Recently, Hinteregger [1979] reported observations of solar cycle 21 which were made continuously from July 1976 to January 1979 and are summarized in Figure 18. His results indicate a factor of 2.8 variation at Lyman-alpha but only a 20% increase in solar flux at 1800 Å. Hinteregger's observations have been extrapolated to 2000 Å and are assumed to represent the variation in solar flux between the

Fig. 16. Seasonal variation of ozone. The relative seasonal variation of ozone at mid-latitudes from the ROCOZ data (dots) is compared with results from the standard photochemical model with 30°N seasonal atmospheres (solid lines). The annual means are shown in Figure 15. Typical errors for the data points are given.

Fig. 17. BUV albedos at five wavelengths for specific solar zenith angles. For each of the five standard BUV wavelengths, the mean AE-E albedos are shown for solar zenith angles from 10° to 89° (dots connected by line). The mean tropical Nimbus-4 albedos are given for zenith angles of 10° and 30° only (triangles). BUV albedos have been calculated for the observed ozone profiles from ROCOZ (crosses), from Hilsenrath [1971] (squares), and from solar occultation with the AE-E instrument [Guenther et al., 1977] (circles). These albedos are shown only for the relevant wavelengths and the zenith angles. The albedos from the standard photochemical model (Table 1) for a range of H2O mixing ratios (2-5 ppm) are given for all wavelengths at all angles (vertical bars).

The predominant effects at solar maximum are due to the increased solar radiation at Lyman-alpha and have little effect on the ozone concentrations below 60 km. At solar maximum the odd hydrogen concentrations are greatly enhanced (Figure 7), and ozone is depleted by 25 and 50% at altitudes of 70 and 80 km, respectively. Some caution is necessary since the greater rate of H2O photolysis will enhance the conversion of H2O to H2. Depending on the rate of vertical transport [Keneshea et al., 1979; Allen et al., 1980], the H2O concentration at 80 km may fall below 3 ppm, thus canceling the predicted ozone depletions. Assuming fixed H2O profiles, the model predicts greater albedos at solar maximum, at most 1.5% for the shorter wavelengths near sunset. The effect for solar zenith angles less than 40° would be below 0.5%. Thus changes in the solar flux associated with the 11-year solar cycle do not appear by themselves to produce a detectable variation in ozone

Fig. 18. Variation in ultraviolet solar flux. The ratio of solar flux near solar maximum (January 1979) to that at solar minimum (July 1976) has been observed by Hinteregger [1979]; the observations at Lyman-alpha and from 1400 to 1800 Å (solid line) have been extrapolated to 2000 Å (dashed line). The amplitude of short period (10 day) changes is indicated by the dotted line [Hinteregger, 1979].
shown for only 30° and 86°, the ratio of albedos is a smoothly varying function of zenith angle from 10° to 89°. The standard model with \( k_{17} = 1 \times 10^{-10} \text{ cm}^3 \text{s}^{-1} \). The BUV albedos used in this table are the average of the morning and afternoon albedos. While r.m.s. variation in the monthly mean BUV radiances places an uncertainty in the monthly mean BUV radiances places an upper limit both on the natural variation of ozone with longitude and time and on the short-term variability in solar flux between 2555 and 2922 Å.

### h. Detailed Comparison with Theory

A detailed comparison of the Nimbus-4 albedos with those from two theoretical models is shown in Table 6. The standard photochemical model appears to underestimate the quantity of ozone and thus overestiimates the BUV irradiances by 9–16%. The dry theoretical model provides much better agreement but is still 5% greater than the observed albedos. The Nimbus-4 2555 Å albedo is anomalously higher than the other wavelengths when they are compared with the dry model. In view of the strong overlap in regions sampled by the BUV (Figure 8), the 5% excess at 2555 Å is difficult to explain. A possible explanation could be the occurrence of an error of equal magnitude in the ozone cross section at 2555 Å (Table 4), but the cross section at this wavelength has been measured several times [Griggs, 1968].

Resonance fluorescence by nitric oxide has been proposed as a source of contamination in the 2555 Å BUV albedos [Guenther et al., 1979]. A substantial fraction of the sunlight at 2150 Å is reradiated in the \((v', v'') = (1, 4)\) gamma band near 2555 Å. Guenther et al. [1979] note that the major contribution is from thermospheric NO and that it may be as large as 5% of the BUV signal at high latitudes. In the tropics, however, thermospheric NO densities are typically 5 times less than those observed at high latitudes [Cravens and Stewart, 1978]. Thus the nitric oxide contribution to the tropical 2555 Å radiances should be less than 1% for daytime observations.

At this point we can compare the ROCOZ data from 38°N with the Nimbus-4 tropical data. Both sets of observations present a consistent pattern which is well described by a standard photochemical model with 2 ppm H₂O or, equivalently, with \( k_{17} = 1 \times 10^{-10} \text{ cm}^3 \text{s}^{-1} \). This model reproduces to within 6% by the theoretical models. The standard photochemical model matches the diurnal behavior of the shorter wavelengths, whereas the dry model is more representative of the longer wavelengths. One possible conclusion is that from 50 to 60 km either the H₂O mixing ratio increases from 2 to 5 ppm or else the rate constant for OH + HO₂ (R17) decreases by the same factor. The anomalous period January 1976 is consistent with a much wetter stratosphere (~8 ppm H₂O).

The asymmetry in the daytime ozone concentrations shown in Figure 6 leads to a morning-afternoon difference in the backscattered ultraviolet which should be observed by satellite. In Figure 19 the modeled ratio of afternoon-to-morning backscattered light is compared with values from the three periods of AE-E data. Potentially large, systematic shifts in these ratios can be caused by an ephemeris error as small as 3 s. The model predicts less than 1% difference between the corre-

### TABLE 5. Ratio of AE-E BUV Albedos to Photochemical Model Albedos at Solar Zenith Angles of 30°/86°

| Wavelength, Å | January 1976 | July 1976 | January 1977 | January 1976 | July 1976 | January 1977 |
|---------------|---------------|-----------|---------------|---------------|-----------|---------------|
| 2555          | 0.58/0.63     | 0.53/0.56 | 0.57/0.59     | 0.64/0.72     | 0.60/0.64 | 0.63/0.67     |
| 2735          | 0.61/0.65     | 0.59/0.60 | 0.61/0.61     | 0.68/0.74     | 0.65/0.68 | 0.67/0.69     |
| 2830          | 0.64/0.67     | 0.62/0.62 | 0.63/0.62     | 0.70/0.75     | 0.68/0.70 | 0.68/0.69     |
| 2876          | 0.66/0.67     | 0.64/0.63 | 0.64/0.62     | 0.71/0.75     | 0.69/0.70 | 0.69/0.69     |
| 2922          | 0.69/0.67     | 0.67/0.64 | 0.66/0.62     | 0.73/0.73     | 0.72/0.70 | 0.71/0.68     |

The models are described in Table 5. The Nimbus-4 BUV albedos are from the periods June–August 1970 and December 1970 to February 1971. The values represent the mean of ratios at solar zenith angles 10° and 30°. The typical month-to-month variation in the mean Nimbus-4 albedo at 30° zenith angle is ±2%.

### TABLE 6. Ratio of Nimbus-4 BUV Albedos to Photochemical Model Albedos

| Wavelength, Å | Standard Model | Dry Model |
|---------------|----------------|-----------|
|                | July 1970 | January 1971 | July 1970 | January 1971 |
| 2555          | 0.87     | 0.91      | 0.98     | 1.02      |
| 2735          | 0.85     | 0.86      | 0.94     | 0.95      |
| 2830          | 0.86     | 0.88      | 0.93     | 0.95      |
| 2876          | 0.88     | 0.90      | 0.95     | 0.97      |
| 2922          | 0.88     | 0.90      | 0.94     | 0.96      |

The models are described in Table 5. The Nimbus-4 BUV albedos are from the periods June–August 1970 and December 1970 to February 1971. The values represent the mean of ratios at solar zenith angles 10° and 30°. The typical month-to-month variation in the mean Nimbus-4 albedo at 30° zenith angle is ±2%.
sponding morning and afternoon albedos at 2922 Å; thus, the 2922 Å albedo has been used to locate precisely the time of observation (e.g., the 2876 Å observation occurs 2 s after 2922 Å and so on). The AE-E data have the same P.M./A.M. patterns as the photochemical model and one cannot readily distinguish between models with 2 to 5 ppm H₂O. The P.M./A.M. ratios from January 1976 are indistinguishable from the other periods.

The AE-E observations point to an annual cycle in the 2555 Å albedo, notably at large zenith angles. There is a January maximum and July minimum at short wavelengths which is only partially explained by the model through seasonal changes in the background atmosphere [Cole and Kantor, 1978]. The reduced albedo during the months of June, July, and August appears also in the Nimbus-4 data; however, the pattern may not extend to July 1971 (P. K. Bhartia, private communication, 1980). This indication of a July maximum in the ozone absorption above 40 km differs from the mid-latitude results of Deluisci et al. [1979]. They analyzed inverted ozone profiles for the same period of Nimbus-4 observations and infer a summer minimum in ozone (pressure versus pressure) above 40 km for 36°N to 52°N. The mid-latitude rocket observations (38°N) show everywhere a summer minimum in ozone partial pressure versus total pressure above 60 km. For modest activation energies [cf. Logan et al., 1978], the resulting increase in ozone could explain half of the observed excess in the annual cycle. As was noted before [Guenther et al., 1979], resonance fluorescence of nitric oxide contributes to the observed intensities at 2560 Å; the 5% annual variation would correspond to a yearly cycle of order 4 x 10¹⁵ NO cm⁻² above 50 km. Since thermospheric NO in the tropics is observed to vary by only 1 x 10¹⁴ cm⁻² [Stewart and Cravens, 1978], such a cycle would require unusually large variations (∼100 ppb) in mesospheric NO [cf. Hudson and Reed, 1979, p. 173].

A major uncertainty in the photochemical model for ozone centers on the kinetic data for the reaction OH + HO₂ (R17). Measurements at low pressures [Burrows et al., 1977; Chang and Kaufman, 1978] give k₁₇ = 4 x 10⁻¹⁰ cm³ s⁻¹, while the reaction appears to be much faster at 1 atm [DeMore, 1979], k₁₇ = 1 x 10⁻⁸ cm³ s⁻¹. Excellent modeling of the ozone observations is achieved within this range of kinetic measurements. The ozone data indicate a possible transition in k₁₇ between 50 and 60 km (0.7–0.3 torr). Similarly, an acceptable model of mesospheric CO observations is possible with the lower value of k₁₇, near 80 km [Allen et al., 1980]. The sensitivity of the photochemistry to the abundance of H₂O makes mesospheric water measurements an essential part of any refinement in the
theoretical model. Such observations must be devised not only to determine the mean profile of water vapor but also to detect annual short term variability, for example from meteoritic sources.

The photochemical lifetime of \((O + O_2)\) is less than 1 day between 40 and 80 km. Thus the response of ozone to changes in the composition and structure of the atmosphere will be rapid. Global monitoring of ozone in the upper stratosphere and mesosphere provides information not only on long-term trends in the concentration of ozone but also on rapid changes in the composition and structure of the upper atmosphere. To interpret the latter, we must increase our confidence in the photochemical modeling of ozone through continued in situ measurements of ozone by rocket and balloon.

Observations of ozone in the upper stratosphere and mesosphere on a global scale are best accomplished by satellite BUV observations at selected wavelengths between 2200 and 2950 Å. The instrument should have a narrow field of view \((1°-3°)\) and be placed in a circular orbit below 800 km. The orbit must include timely coverage of the full range of local solar zenith angles at a given latitude. The geometric simplicity of nadir observations does not place stringent requirements on the pointing accuracy, makes the analysis of large data sets straightforward and also allows for rapid observations \((1°)\) of small-scale horizontal structures \((8\, \text{km})\) in the atmosphere. Such an instrument must also be able to take advantage of solar and stellar occultations by the atmosphere in order to measure twilight and nighttime ozone profiles.

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