Current Challenges toward In Vitro Cellular Validation of Inorganic Nanoparticles

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ABSTRACT: An impressive development has been achieved toward the production of well-defined “smart” inorganic nanoparticles, in which the physicochemical properties can be controlled and predicted to a high degree of accuracy. Nanoparticle design is indeed highly advanced, multimodal and multitargeting being the norm, yet we do not fully understand the obstacles that nanoparticles face when used in vivo. Increased cooperation between chemists and biochemists, immunologists and physicists, has allowed us to think outside the box, and we are slowly starting to understand the interactions that nanoparticles undergo under more realistic situations. Importantly, such an understanding involves awareness about the limitations when assessing the influence of such inorganic nanoparticles on biological entities and vice versa, as well as the development of new validation strategies.

INTRODUCTION

The use of nanoparticles (NPs) for drug delivery and imaging is undoubtedly one of the most important areas in biomedicine.1–3 This relatively new field, known as nanomedicine, merges distinct disciplines such as chemistry, pharmacology, immunology, and even electronics for applications such as biomolecular sensing. One of the central features in nanomedicine is the controlled interaction of NPs with target cells,4–7 in such a way that physical and chemical obstacles are overcome, while avoiding undesired toxicity in the long term.8 We are currently seeing a renewed interest in studying how the intrinsic properties of nanomaterials are related to the results we see in vivo.9–11 Consequently, we are asking again all the important questions as to why nanomaterials are failing clinical trials in such high numbers? How do the physicochemical features of NPs change when they are suspended in biological fluids?12 Can cell–NP interactions be predicted if protein corona formation is modulated on demand?13 How do NPs act in flow environments, as compared to nonflowing cell cultures? Is mitochondrial activity a suitable read-out for cell viability?14 Addressing such questions has turned a page in our understanding as to why so many NP formulations fail clinical trials.

We focus this Topical Review specifically on inorganic NPs for a number of reasons. NPs are used for biomedical applications because their small size is favorable for different administration routes and allows delivery of active molecules to subcellular locations via various internalization mechanisms. Additionally, the high surface-to-volume ratio of NPs facilitates the incorporation of multiple moieties, such as antifouling or targeting molecules, toward the assembly of multifunctional NPs. While both inorganic and organic NPs share these size-dependent features, it is mainly inorganic NPs that exhibit novel physical properties at the nanoscale, such as localized plasmon resonances, fluorescence, or superparamagnetism, as compared with their bulk or micron-sized counterparts. These features can be exploited in many potential applications regarding imaging, sensing, and drug delivery. In contrast, there are fewer examples of organic NPs (e.g., perylene based nanocrystals) exhibiting such size dependent physical properties.15,16 In inorganic NPs, physical properties can be tailored on demand by modifying the composition, size, or shape, thereby obtaining “responsive” materials toward external stimuli, including magnetic fields or light. These modifications are not easily achieved with organic nanocrystals. In this context, gold NPs can be produced in various sizes and shapes, which determine their optical response (due to localized plasmon resonances); such NPs have been widely exploited for photoacoustic detection, fluorescence, hyperthermia, or surface-enhanced Raman scattering (SERS).17 Another typical example of inorganic NPs used in nanomedicine is iron oxide NPs which can be used as contrast agents in magnetic resonance imaging (MRI) or heat producers for hyperthermia.18 Iron oxide nanoparticles aside, the presence of inorganic NPs in clinical trials is becoming commonplace and it is clear that other inorganic NPs will likely soon enter the clinic.19 Finally, due to this interest in the use of inorganic NPs for clinical applications, we find ourselves in a situation lacking internal controls relating

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to cytotoxicity, dosing, administration protocols, and other aspects such as in vitro models. Equally important is to understand the fate of internalized inorganic NPs (see, for example, a recent study by Wilhem et al. focused on iron oxide NP degradation) and potentially overlooked allergy formation against inorganic NP core components. Herein we thus discuss recent work pointing out the challenges involved in predicting the interactions between inorganic NPs and biological surfaces due to their modifiable physical properties, and the choice of appropriate protocols for in vitro validation on the efficient application of nanomaterials in biomedicine.

**NANOPARTICLES, A WOLF IN SHEEP’S CLOTHING? UNDERSTANDING UNEXPECTED TOXICITY AND COMMON PITFALLS**

The toxicity of inorganic NPs is largely due to alterations in the physicochemical properties of the NPs in biological fluids, and while comparative studies in which changes in NP size, charge, surface chemistry, or the like are investigated, or different inorganic NPs with similar physicochemical properties are compared, discrepancies are continually encountered (see Figure 1). In addition to a lack of standards in the field, common issues encountered in toxicity testing include attempts to compare unrelated types of inorganic NPs, different administration protocols (a problem that is also often overlooked in in vivo studies), poor choice or differences in chosen cell types (resulting in differences in growth or endocytosis kinetics), and frequent lack of nanomaterial stability testing or poor choice of sterilization methods, both of which are a key aspects to any pharmaceutical product. In this context, the toxicity studies by Manshian et al. using quantum dots (QDs) with similar physicochemical characteristics have shown the following: (i) cell-type differences, (ii) differences in the QD agglomeration degree depending on the amount of serum proteins, (iii) differences due to the exact composition of cell culture media, (iv) differences due to varying exposure time, and (v) higher uptake levels not necessarily correlating with higher toxicities. The quantification of cellular uptake is a key factor when determining the toxicity of NPs. Many techniques such as flow cytometry and fluorescence microscopy with poor z-resolution cannot differentiate between NPs adsorbed on the cell membrane and those which have been internalized. Furthermore, most techniques rely on quantifying the proportion of labels on the NP, i.e., fluorescent or radioactive markers which are assumed to remain conjugated to the NP. Important concerns include dissociation of the label from the NP, which may lead to altered biodistribution and/or enhanced toxicity arising from the various individual components. Ironically, often the major concern to the researcher is how to show that the marker remains with the NP, rather than the effectiveness of the NP itself. In order to study NP cellular uptake, novel strategies based on mass spectrometry are being applied for accurate quantification of the NP core uptake, which can be combined with spectral imaging of whole tissues. Another often overlooked aspect when assessing toxicity is the correct choice of the toxicity assay; arguably the most

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**Figure 1.** NPs face significant physicochemical changes upon exposure to biomolecules found in physiological fluids. These effects may be (1) uncontrolled, such as protein corona formation (a), NP aggregation (b), NP dissolution (c), and even removal or exchange of surface ligands (d); or (2) highly controlled, as in the case for ligand–receptor mediated agglomeration of spherical NPs coated with the carbohydrate disaccharide lactose (Lac) triggered by the presence of the protein β-galactoside binding lectin galectin-3 (Gal-3). Physicochemical changes can affect cellular uptake. For example, differences in protein structure (native vs denatured) within the protein corona affect the cellular internalization pathways as shown in illustration 3. Reproduced in part from ref 24 with permission from Elsevier, and from refs 7,37 with permission from the American Chemical Society.
commonly used assays when analyzing the effect of NPs in vitro are cell viability assays based on membrane leakage or mitochondrial activity. With the increasing variety of NP types, biotech firms have worked hard to produce more varied options for measuring cytotoxicity, in which the inference of the NP itself is reduced. Considering spectroscopic tetrazolium based assays, in addition to the demonstration of direct and indirect inhibition of the assay components, a vast majority of inorganic NPs absorb in the measurement window of 400−600 nm, meaning that there is a major risk that the observed cytotoxic effect is incorrectly lower due to absorption by the NP. It has been reported that carbon nanotubes can quench the fluorescence readout in the resazurin assay, and that resazurin can be reduced by molecules such as ascorbic acid which are often used for gold NP synthesis and stabilization. Care must also be taken in understanding what the cell viability assay is truly showing. For example, the Live/Dead fluorescent assay is very useful at providing a visible overview of cell viability based on the membrane integrity of the cells in question. However, cells may have altered membrane integrity yet remain alive, for example, after certain laser treatments. This is especially seen when studying cellular redox levels, as while inorganic NPs can be used to determine intracellular levels of reactive oxygen species (ROS), they are also known to cause ROS production and have been shown to interfere in ROS detection. Cytoskeletal changes, alterations of intracellular signaling pathways, triggering and inhibition of protein fibrillation, and alteration of protein or gene expression are further examples of reported effects mediated by inorganic NPs. In this context, we know that the phenotype of cells can be affected by the presence of inorganic NPs, without directly affecting cell viability, which can result in altered cellular functions such as triggering signaling cascades or distinct inflammatory functions. We refer the reader to several review articles which address in detail common pitfalls while assessing NP cytotoxicity, with special reference to intracellular changes.

FIRST COME, FIRST SERVED? FORESEEING NANOPARTICLE INTERACTIONS WITH THE BIOLOGICAL INTERFACE

The importance of NP shape, size, and surface charge for efficient cellular uptake is well-known (see Canton for a critical review), and recent works aimed at directly investigating such factors with a high degree of control. For example, it is well-known that protein binding to NPs may affect their overall size and charge, and situations in which NPs must cross physical barriers such as the blood-brain barrier or the blood vessel endothelium to enter tumors, must be carefully considered. Protein corona formation depends on the nature of the NPs (composition and physicochemical properties), their surface coating and their route of administration into the body or onto a cell culture. Interestingly, it has been demonstrated that differently charged NPs can affect the molecular structure of the proteins binding to their surface, ultimately affecting cellular uptake (see Figure 1). Therefore, one should be aware that heat inactivation of serum, thereby denaturing serum proteins, in cell culture may also have significant effects on NP uptake. It has been recently demonstrated that the suborgan biodistribution of gold NPs is also affected by NP surface charge and their subsequent interactions with biomolecules. The authors propose neutral NPs as immunologically stronger NPs due to their accumulation in the immune active white pulp of the spleen, possibly due to interactions of the NP surface with proteins such as IgG or fibronectin, the latter known to bind PEG stabilized NPs.

In line with the increased interest in 3D cell models (see below) and enhancing NP uptake via nonspecific and relatively
Review

In cancer cell cultures, the enhanced metabolism of cancer cell lines compared to “healthy” cells can be exploited to achieve higher levels of NP uptake. In vivo, the simplest method to improve NP localization in cancerous tissues involves taking advantage of the enhanced permeability and retention (EPR) effect; however, without NP ligand-cell receptor targeting, this method does not necessarily result in increased levels of cellular uptake (Figure 3). The principle downfall in this strategy is, however, the inherent diversity in cancers, ranging from variations in the EPR effect within the same tumor, cancer location, stage progression, and even between patients. Importantly, after leaking into tumor sites, NPs are known to accumulate in the cells they first encounter, thereby limiting their penetration and potential usefulness. A method to overcome this problem has been demonstrated in vitro by Setyawati et al., using endocytosed nanodiamonds to increase the EPR effect by inducing intracellular ROS production within endothelial cells. The result was a decrease in the intercellular adherence properties and cytoskeletal remodeling, leading to a further increase of NP penetration and allowing the movement of model cytotoxic drugs across the normally nonleaky endothelial cell barrier (Figure 3). Another interesting method to increase NP cytotoxicity in cancer cell models involves predictable intracellular NP aggregation states. Hu and colleagues produced “smart NPs” that take advantage of the relatively high intracellular glutathione concentration to induce reduction of diselenide bonds within their surface coating, leading to high NP uptake of small NPs, massive intracellular flocculation, ROS production, and apoptosis of MCF-7 and HeLa cancer cells in an in vitro model. Intracellular NP aggregation states can also be exploited for imaging or treatment modalities. For example, the photothermal heating efficiency due to plasmon-coupling between gold NPs can be tuned through the size and morphology of NPs, as well as their aggregation state within intracellular endosomal compartments. Equally interesting are methods to avoid intracellular NP aggregation, such as the encapsulation with silica or amphiphilic polymer shells, thereby ensuring that the optical properties of gold NPs remain for applications such as photothermal therapy or imaging.

MOVING FORWARD: 3D TISSUE CULTURE SYSTEMS, MICROFLUIDICS, AND IN SILICO SIMULATIONS

Numerous techniques are under development for the characterization of the interactions of NPs with eukaryotic cells in an environment that resembles the real situation in vivo. Such techniques can be divided into two basic groups, those addressing the interactions of cells with each other and with external supports, and those that focus on the flow that is present in vivo to allow the diffusion of both soluble and insoluble factors. Within the first group, current research focuses on moving from 2D to 3D, and from single cell to multicell type (co)cultures. Therefore, the interaction between inorganic NPs and cells is starting to take into consideration (among other facts) the existence of extracellular barriers, multiple cell type interactions, and possible modifications in the cell phenotype due to interactions between cells and their culture dish supports. The simplest models are acellular or cell-seeded gels, which are useful to study NP transport, uptake, and therapeutic efficiency. However, their use is limited because particles can interact with the gels but cannot easily mimic tissue structures. Interestingly, multicellular spheroids, which do not require an external scaffold for cell packing, or multilayer cell cultures, are more advanced systems to study NP tissue penetration, targeting, and toxicity. Several studies have aimed to compare the penetration, toxicity, and targeting of various kinds of inorganic NPs between multicellular spheroids and animal models. There appears to be a good correlation between in vitro 3D and in vivo models, showing the usefulness of such 3D coculture models to comply with the 3R’s and overcome ethical issues related to animal testing. A recent study using stem cell spheroids and magnetic NPs has elegantly shown the long-term effects of both NPs on cells, and vice versa, i.e., what is the extent of endosomal iron oxide NP degradation and how does free iron affect gene expression. Commercially available 3D spheroids can help to standardize protocols to study NP–cell interactions. However, such models have limitations mainly related to the nature of the spheroid itself, which cannot match the level of complexity of true tissues as well as vascularization. Notwithstanding, technologies based on stem cells have opened...
new opportunities for in vitro experimentation. It is now possible to fabricate miniaturized organs, so-called organoids, that can replace in some cases ex vivo systems and open up new horizons in human biology research, overcoming some of the limitations involved in using animal models. Such promising 3D coculture models include organoids mimicking the intestine, breast, brain, kidney, heart, and lungs. They are essentially aggregates of several cell types fabricated from pluripotent stem cells, adipose-derived stem cells and tissue progenitors from animal or human sources that are able to form some of the complex structures of organs, self-organize, self-renew, and perform cellular functions typical from in vivo tissues. Therefore, they can be used to study nutrient transport, tissue replacement therapy, disease diagnosis, drug screening, or toxicity and be applied for personalized medicine. Interestingly, they are bridging the gap between in vitro and in vivo experimentation. Recently, two detailed protocols describing the production of liver, kidney, and pancreas organoids have been published, which may be crucial in the field of nanomedicine to assess NP cytotoxicity or drug release, for example. In this context, Astashkina et al. studied the cytotoxicity and cytokine production of gold and polymeric NPs in a 3D kidney organoid, observing that only the organic NPs were able to penetrate the organoid and produce indicators of toxicity. While we are convinced that this exciting and emerging research field will raise the number of studies assessing inorganic NP−cell interactions and NP therapeutic efficiency, a lack of standards and “teething” issues such as the lack of nutrients and oxygen at the center of the organoid, resulting in cellular necrosis, limits current use for NP validation.

All the previously mentioned 3D coculture models lack vascularization and do not consider that NPs in vivo are immersed in a fluid flow. Such flows affect many aspects of cell−NP interactions, ranging from tissue biodistribution through stress forces at the cell surface−NP level. In the most simple context, gravity affects NP sedimentation rates and subsequently NP uptake by cells. Microfabrication and microfluidic systems have the potential to contribute greatly in this field. Basic perfusable 3D culture models exist that allow control over flow rate, pH, or drug concentration, in some cases permitting in situ visualization. Ng and Pun studied and compared the penetration and uptake of fluorescent NPs of two different sizes under interstitial fluid flow in a perfusable 3D monoculture of cells. It was possible not only to image the penetration in situ but also to detach the cells from the scaffold and study the NP uptake using flow cytometry. Results were similar to those obtained with multicellular spheroids under static conditions. Indeed, the most advanced perfusable 3D cell culture systems to study cell−NP interactions, NP penetration, and drug delivery are currently based on tissue-on-a-chip platforms, and the more advanced organ-on-a-chip technology is being used for drug screening. In parallel to the development of microfluidic devices including multi-cell type cultures for in situ imaging, a pool of experimental data derived from animal models will be essential for comparison (Figure 5). This will shed light on the open question of whether it will be possible to
truly correlate results obtained with in vivo animal models and advanced in vitro models.\textsuperscript{109}

In silico simulations that can predict in vitro inorganic NP uptake and intracellular distribution, as well as in vivo NP fate and drug delivery efficiency, are not yet commonplace.\textsuperscript{115,116} We however expect that they will become increasingly important toward the optimal design of drug carriers, prediction of NP uptake, pharmacokinetics, toxicity, and determination of biodistribution within cells of targeted tissues.\textsuperscript{81,117,118} For example, computational fluidic dynamics is used to understand which aerodynamic and physicochemical factors of NPs affect their circulation in blood.\textsuperscript{119,120} In addition, simulations have also provided information regarding the way NPs extravase out of blood vessels, diffuse, and reach targeted cancer cells. Combination of this research with experimental data will be a powerful tool for the implementation of nanomedicine based on “safe-by-design” production of NPs. However, collaborative efforts between scientists are required to face the challenges that exist in modeling NPs in biological environments in a more realistic manner.\textsuperscript{121}

\section*{CONCLUSIONS}

The future of NPs in drug delivery and imaging will rely on the ability to produce standardized protocols by which we can compare our findings on aspects ranging from physicochemical properties and stability to in vitro and in vivo performance. Unfortunately, the wide variety of inorganic NP cores and surface coatings used to produce drug delivery or imaging systems complicates this aspiration. Although we are increasingly able to better understand why NPs behave as they do, the time and effort required for the validation of a new NP formulation is not yet efficient; essentially, new methods for the prediction of NP interaction with living organisms are required. Like most pharmaceutical products, inorganic NPs are not perfectly stable, especially in biological fluids, and their aggregation state and surface properties including the formation of protein corona make their characterization complicated.

\begin{figure}
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\caption{Current methods to predict in vivo responses rely on the toolbox of available in vitro techniques, ranging from 2D to 3D cultures, involving single or multiple cell types (1).\textsuperscript{109} An example of a perfusable 3D cell culture designed as a spheroid-on-a-chip.\textsuperscript{112} The microfluidic device (A) contains a cancer cell spheroid embedded in an extracellular matrix (B), which permits real-time optical analysis of the penetration of NPs with different sizes (C). Reproduced in part from ref\textsuperscript{109} with permission from Elsevier, and from ref\textsuperscript{112} with permission from Nature Publishing Group.}
\end{figure}
While methods emerge to standardize in vitro testing and to fabricate systems better resembling the in vivo environment, additional efforts to provide standards and comparative methods in NP production are required. We should remember that nanomedicine is a relatively new field involving cross-disciplinary scientists, and while a large number of inorganic NPs have advanced from the lab to the clinic, collaboration is key. New technologies for in vitro testing and new computational methods to realistically predict the interaction of NPs with living organisms starting from cellular models will be very important. We also predict an increase in collaborations between materials scientists and medical practitioners, thereby addressing from the start what are the important pharmaceutical goals and the current flaws.

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**Notes**
The authors declare no competing financial interest.

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