Meat Quality, Blood Profile, and Fecal Ammonia Concentration of Broiler Supplemented with Liquid Smoke

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ABSTRACT

Liquid smoke is one of the feed additive that can be given to animals. The aim of the study was to determine the influence of liquid smoke given through drinking water on meat quality and production of fecal ammonia in broiler. Variables observed were meat pH, water holding capacity, cooking loss, the tensile strength of meat, cut off strength of meat, blood urea nitrogen and creatinine concentrations, and fecal ammonia. The experiment was assigned in a completely randomized design with 5 treatments, and 4 equal replicates. The treatments tested were symbolized as R0, R1, R2, R3, and R4, based on the level of liquid smoke added into drinking water (v/v) namely, 0%, 0.25%, 0.50%, 0.75%, and 1.00%, respectively. The data were subjected to analysis of variance, and continued to Duncan’s multiple range test to determine the difference between treatment mean values at 5% probability. The results indicated that addition of liquid smoke up to 1% did not affect the meat pH, water holding capacity, cooking loss, the tensile strength of meat, blood urea nitrogen, and creatinine content, but it decreased the cut off strength of meat and fecal ammonia. It was concluded that an optimal dose of granting liquid smoke through drinking water was 1%.

Key words: liquid smoke, broiler, meat quality, fecal ammonia

INTRODUCTION

Additive is one of the feed ingredients that commonly used in the formulation of animal feed, since it has a very important role in increasing the productivity and health of poultry (Sofyan et al., 2010). One of the potential additional feed given to livestock is liquid smoke (Wang et al., 2012). Liquid smoke is a mixture of aqueous dispersion of wood smoke in the water created by the process of condensation of wood smoke resulted from pyrolysis at a temperature of 600-900 °C (Wang et al., 2012). Liquid smoke is generated from wood that is roughly comprised of 50% cellulose, 25% hemicellulose, and 25% lignins (Simko, 2005). Liquid smoke contains carboxyl acid compounds produced by the decomposition of hemicellulose during pyrolysis (Akakabe et al.,

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Birds, Feed, and Cage

A total of 120 unsexed Cobb broiler chicks bought from commercial poultry in South Sumatera, Indonesia were used in the current study. Chicks were fed commercial starter diet (crumble) from 7-21 d of age, and commercial finisher diet (pellet) from 22-35 d of age. The commercial feed was purchased from poultry shop. The nutritional contents of the feed used in the present study were shown in Table 1. Liquid smoke was obtained from Badja Baru factory located in Palembang, South Sumatera. The content of compounds in liquid smoke used in this study was shown in Table 2. All chicks were reared in a conventional open-sided house by allocating into 20 units of cage of 70 x 70 x 60 cm (l x w x h) in size with 6 birds each.

Experimental Procedure

Liquid smoke was provided through drinking water and given every day from 1-35 d of age. The treatments tested were R0, R1, R2, R3, and R4. The first symbolized as R0 function as a control (without added liquid smoke), while the other four namely, R1, R2, R3, and R4 were treatment with the addition of liquid smoke into drinking water with the level of 0.25%, 0.50%, 0.75%, and 1.00%, respectively. Level of 0.25% was equal to 2.5 mL liquid smoke per one liter water.

Table 1. Nutrient contents of experimental diet (dry matter basis)

| Nutrient contents | Starter diet (7-21 d of age) | Finisher diet (22-35 d of age) |
|-------------------|-----------------------------|-------------------------------|
| Crude protein (%) | 21                          | 19                            |
| ME (Kcal/kg)      | 3000                        | 3100                          |
| Fat (%)           | 4                           | 4                             |
| Fiber (%)         | 5                            | 5.5                           |
| Ash (%)           | 6                            | 6                             |
| Calcium (%)       | 0.9                          | 0.9                           |
| Phosphorus (%)    | 0.7                          | 0.7                           |

Source: PT. Japfa Comfeed

Table 2. Compound contents of liquid smoke used in this study

| Compounds                        | Time of retention (minute) | %     |
|----------------------------------|----------------------------|-------|
| 1-Methyldiane                    | 3.554                      | 0.66  |
| 1,3-Dimethylbenzene              | 3.628                      | 0.43  |
| Phenol                           | 3.674                      | 0.39  |
| 4-amino                          | 3.746                      | 3.58  |
| 1,3-Dimethylbenzyl              | 3.891                      | 0.74  |
| Cevane-3, epoxy, 3-acetate       | 4.115                      | 5.72  |
| Phenol, 2-Methoxy               | 4.286                      | 5.65  |
| 2-Isopropenylthiophene           | 4.349                      | 3.17  |
| Cyclohexane                      | 4.504                      | 2.57  |
| Phenol                           | 4.589                      | 3.73  |
| 3-fluoro-o-xylene                | 4.801                      | 4.95  |
| Imidazole-5-carboxylic acid      | 5.516                      | 2.03  |
| 1,4-Benzenedimethanol            | 6.277                      | 1.15  |
| Phenol, 2-methoxy-4-methyl       | 6.355                      | 3.11  |
| Cyclobutene                      | 8.086                      | 13.16 |
| Pentalenone                      | 8.732                      | 6.26  |
| 1,2-Benzenediol                 | 8.869                      | 1.85  |
| 1-pentenalone                    | 8.95                       | 6.26  |
| Phenol                           | 11.313                     | 16.3  |
| Benzoic Acid                     | 13.911                     | 6.65  |
| Benzene, 1,2,5-trimethoxy-3-     | 15.965                     | 2.91  |
| methyl                           |                            |       |
| 3-Furancarboxylic acid           |                            |       |
| 2,5-dihtydroxy-4-isopropyl-2,4,6-cy | 16.285                  | 1.89  |
| Benzenecet酸 acid                |                            |       |

Note: Sample was analyzed in Lab. of Regional Health, Central Jakarta.
VariablesMeasured

At the end of experiment (35 day of age) one bird from each replication was randomly decapitated for meat sampling. The breast muscles (without skin) were used for physical meat quality analysis (pH, cooking loss, water holding capacity, tensile strength, and cut off strength). The meat samples were analyzed about 1-2 h after decapitation.

pH value. Meat pH was measured according to Soewedo (1994). Meat samples were mashed with meat grinder, and 2 g sample was diluted with 18 mL of distilled water and stirred until homogen and then filtered. The filtrate samples were measured with a pH meter.

Cooking loss. Cooking loss was determined according to Nikmaram et al. (2011). Meat sample (20 g) was placed in polyethylene plastic, then sealed with vaccum pack, and heated in a water bath at 80 °C for 30 min. After being cooked, samples were cooled at room temperature, dried surface with filter paper, reweighed by using an analytical balance (Metler AE100-0.001), and the cooking loss was calculated from difference between raw and cooked weights.

Water holding capacity. Water holding capacity (WHC) was determined according to Hamm method (1960). Meat sample (0.3 g) was put on Whatman 41 filter paper and placed between two metal plates with a pressure load of 35 kg for 5 min until wet area was formed on the filter paper. Wet area was calculated by subtracting the total area (wide outer ring) with the area covered meat samples (circle area).

The strength of tensile and cut off of meat. The strength of tensile and cutoff meat was measured by the method of Person & Dutson (Murtini & Qomarudin, 2003). This procedure used universal testing instrument merk Lloyd in units of newton. Samples were cut into a cube with a size of 1 x 1 x 1 cm. The samples were laid under a represor and then the instrument was executed.

Blood urea nitrogen and creatinine. Blood urea nitrogen and creatinine were measured by using serum sample. At the end of the study, around 5 mL blood sample was taken from the jugular vein and immediately pro-
cessed at the laboratory of clinical pathology (Salasia & Khusnan, 2001).

Ammonia concentration. Ammonia concentration was measured by the method of micro diffusion Conway. About 1 mL of supernatant was placed on one side of the Conway cups septum placed obliquely toward the septum. The Conway cup was formerly smeared with vaseline on both surfaces. The other side was filled with 1 mL of saturated aqueous Na₂CO₃, while the central part was filled with 1 mL of boric acid. The Conway cup was closed until airtight and shaken so that the super-

Experimental Design and Statistical Analysis

The experiment was assigned in a completely randomized design with 5 treatments (as described in the Section of Experimental Procedure) and 4 equal replications (6 birds each). The data were subjected to analysis of variance, and the differences between the treatments were analyzed by Duncan’s multiple range test (Stell & Torrie, 1980).

RESULTS AND DISCUSSION

Meat Quality

The result showed that the use of liquid smoke through drinking water did not affect the meat pH. The average of meat pH was 5.83 (Table 3). The value of the meat pH on this study was lower than normal pH of broiler meat. Van Laack et al. (2000) stated that the normal pH value of broiler chicken meat was ranged between 5.96 and 6.07. Suradi (2008) stated that the average of early pH of the breast meat on broiler chicken was 6.31, and then it declined after postmortem and reached the values of 5.96-5.82 at 6-10 h. The nonsignificant effects of treatment on meat pH was possibly due to the low total organic acid content of liquid smoke, i.e. 19.2% (Table 2). In this study, organic acids contained in liquid smoke were acetic acid, carboxylic acid, benzoic acid, and furancarboxylic acid. The organic acid has ability to inhibit the process of rigor mortis on meat so that it

Table 3. Quality of broiler chicken meat aged 5 weeks and supplemented with various levels of liquid smoke in drinking water

| Parameters                | Treatments       |
|---------------------------|------------------|
|                           | R0               | R1               | R2               | R3               | R4               |
| pH                        | 5.76±0.14        | 5.73±0.19        | 5.92±0.04        | 5.91±0.03        | 5.82±0.17        |
| Water holding capacity (%)|                  |                  |                  |                  |                  |
| Tensile strength (Newton) | 22.87±1.36       | 22.43±2.30       | 18.93±4.66       | 18.80±3.47       | 22.87±1.76       |
| Cooking loss (%)          | 34.75±3.21       | 35.30±2.16       | 28.83±4.69       | 30.81±5.41       | 27.97±6.24       |
| Cut off strength (Newton) | 0.94±0.27ab      | 0.83±0.14ab      | 0.70±0.15ab      | 0.55±0.04bc      | 0.33±0.09abc     |

Note: Means in the same row with different superscripts differ significantly (P<0.05). R0= drinking water without added liquid smoke (control), R1= drinking water + liquid smoke 0.25%, R2= drinking water + liquid smoke 0.5%, R3= drinking water + liquid smoke 0.75%, R4= drinking water + liquid smoke 1.00%.
maintains the pH of meat remains high. Abustam & Ali (2011) stated that the high acid content in the liquid smoke could slow or delay the process of rigor mortis, so a high meat pH can be retained. In line with this, Suradi (2008) stated that the lower the pH of meat, the quality of meat will also be getting low.

The result showed that the use of liquid smoke through drinking water did not affect the WHC of meat. The average of WHC of meat was 21.58% (Table 3). Soeparno (2005) stated that WHC of broiler meat at the age of 6 and 7 weeks was 22.19% to 28.54%. There were a correlation between WHC and meat pH (Tang et al., 2007), whereas the decrease in meat pH results in the decrease in WHC (Bee et al., 2007; Swatland, 2008; Jung et al., 2010). Alvarado & McKee (2007) stated that one of the factors that affected the WHC of meat was acidity/pH. The organic acids contained in liquid smoke could make a bonding strips of myofibril became looser (Abustam & Ali, 2011). This looser binding then forms the empty space that is eventually filled by water so that the WHC of meat increased. Abustam & Ali (2011) reported that the more liquid smoke used, the WHC of meat was getting low. Due to the meat pH in this study was not significantly different, so that the WHC of meat was not also different. Huff-Lonergan & Lonergan (2005) reported that WHC was also influenced by postmortem proteolysis.

The result showed that the used of liquid smoke through drinking water did not affect the tensile strength of meat. The average of tensile strength of meat in this study was 3.41 Newton (Table 3). There was a positive correlation between the tensile strength of meat and meat pH. Min & Ahn (2005); Ke et al. (2009) stated that the pH of meat was one of the indicators that determined the value of tenderness, where one of them was reflected by the value of tensile strength of meat. Murtini & Qomarudin (2003) stated that when the force required to pull the meat was lower, the tenderness of the meat was increased. In another word, the low value of tensile strength will improve the quality of meat (Utami et al., 2011).

The result showed that the used of liquid smoke through drinking water did not affect the cooking loss of meat. The average of cooking loss of meat in this study was 31.35% (Table 3). Soeparno (2005) stated that cooking loss of broiler meat at the age of 6 and 7 weeks was 24.89% to 34.57%. The cooking loss of meat was correlated with meat pH. If the meat had a high pH, the cooking loss of meat would be high too. The value of cooking loss was used to predict the amount of fluid in cooked meats (Soeparno, 2005). Suradi (2008) reported that the meat with a low cooking loss had a physical quality that was relatively better than the meat with a high cooking loss. This result was related to the lower nutrients lost during cooking.

The cut off strength of broiler meat was significantly (P<0.05) affected by treatments. The highest cut off strength of meat was found in group of R0 (Table 3). The use of liquid smoke as much as 1% through drinking water (R4) significantly (P<0.05) decreased cut off strength of meat to the lowest value (0.33 Newton) and it was significantly different as compared to R0, R1, R2, and R3. The use of liquid smoke in R4 might achieve an ideal condition for cut off strength of meat. Prayitno et al. (2010) stated that cut off strength of meat was affected by meat pH and WHC. In this study, the meat pH and WHC were not significantly affected by liquid smoke but it significantly affected the cut off strength of meat. The difference was possibly due to the time while measuring the cut off strength, whereas the measurement was conducted in the last session. It finally affected the value of meat pH. In the beginning, meat pH was still high and it gradually declined with the increased duration before measurement. This condition was caused by glycolysis process that produced lactic acid in the meat (Park et al., 2010). The decline in the value of the cut off strength of meat is one of the indicators of tenderness. Utami et al. (2011) reported that the lower the cut off strength of meat, then the higher the tenderness of meat.

### Blood Urea Nitrogen, Creatinine, and Ammonia

The result showed that the used of liquid smoke through drinking water did not affect the BUN and creatinine. The average of BUN and creatinine concentrations were 6.76 mg/dL and 0.17 mg/dL, respectively (Table 4). The same results also shown from the research of Yan et al. (2012), that supplementation of liquid smoke did not show the significant effect on creatinine. This study indicated that the use of liquid smoke until 1% through drinking water was still safe to be consumed by broiler chicken and did not disturb the kidney function. Scholz (2005) stated that the levels of BUN and creatinine were very sensitive parameter to describe the function of kidneys. If the glomerulus cells on kidney damaged, the rate of glomerulus filtration became declining. This would finally cause urea and creatinine accumulation in the plasma. In line with this, Wahjuni & Bijanti (2006)

| Parameters                  | R0     | R1     | R2     | R3     | R4     |
|-----------------------------|--------|--------|--------|--------|--------|
| Blood Urea Nitrogen (BUN) (mg/dL) | 6.79±1.42 | 7.20±1.87 | 5.85±2.53 | 7.61±1.33 | 6.35±1.07 |
| Creatinine (mg/dL)          | 0.23±0.23 | 0.07±0.05 | 0.14±0.05 | 0.12±0.07 | 0.27±0.02 |
| Ammonia concentration (ppm) | 115.25±10.03 | 81.71±3.90 | 76.78±2.74 | 65.94±4.43 | 60.62±3.71 |

Note: Means in the same row with different superscripts differ significantly (P<0.05). R0= drinking water without added liquid smoke (control), R1= drinking water + liquid smoke 0.25%, R2= drinking water + liquid smoke 0.5%, R3= drinking water + liquid smoke 0.75%, R4= drinking water + liquid smoke 1.00%.
also stated that the increased concentration of creatinine in the blood indicated the kidney function was getting corrupted. There were also a relationship between protein intake and creatinine (Araujo et al., 2006). Although the protein intake affected creatinine, its influence was not direct because creatinine was synthesized from creatine using essential amino acids, arginine and glycine, as precursors.

The use of liquid smoke through drinking water significantly (P<0.05) affected the amount of ammonia. This results showed that the amount of ammonia given by the treatment of R0 was the highest (P<0.05) among the other treatments. The amount of ammonia produced between treatment R1 and R2 did not differ significantly, but higher (P<0.05) than the treatment of R3 and R4, and lower (P<0.05) than treatment of R0. Meanwhile, the amount of ammonia generated between the treatment of R3 and R4 were not significantly different. Based on Table 4, the highest amount of ammonia produced was in the treatment of R0 (without the addition of liquid smoke through drinking water), which was 115.25 ppm.

The high production of ammonia in excreta showed that the process of proteins digestion in feed did not run optimally. Hendalia et al. (2012) stated that decreased digestibility of feed protein caused the decreased nitrogen absorption and increased nitrogen excretion through excreta in the form of uric acid, and it would finally be transformed into ammonia by bacteria. Based on this study, supplementation of liquid smoke through drinking water could increase the digestion process of the feed proteins in the digestive tract. This was closely related to the role of chemical compounds in the liquid smoke, one is an organic acid. Chen et al. (2010) and Yan et al. (2012) stated that the organic acids in liquid smoke could reduce the release of total kjeldahl nitrogen (TKN) during the process of decaying of excreta, so that it finally could reduce emissions of gas by livestock. In line with this, Asada et al. (2006) also reported that with the granting of bamboo vinegar, it make the release of TKN became more effective.

The amount of ammonia decreased with the supplementation of liquid smoke through drinking water. This result is in line with result reported by Yan et al. (2012) that supplementation of bamboo vinegar could decrease ammonia in pig feces. North & Bell (1990) stated that the amount of ammonia inside the cage should not exceed 25 ppm. If the ammonia levels inside the cage exceed 25 ppm, the productivity of chicken would decrease and it could cause the death in chickens if the level of ammonia reached 0.05% or 500 ppm. In this study, the level of ammonia produced was gradually declined with the increased level of liquid smoke supplementation. This technique was good for chickens to keep them in health.

CONCLUSION

The optimal level of liquid smoke supplementation through drinking water was 1%, because it could decrease the cut off strength of meat and production of fecal ammonia. The value of meat pH, WHC, cooking loss, the tensile strength of meat, BUN, and creatinine content were not affected by the liquid smoke supplementation.

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