Impact of High-Pressure Homogenization on the Extractability and Stability of Phytochemicals

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High-pressure homogenization (HPH) and high-pressure processing (HPP) are emerging technologies for the food industry. Both technologies employ high pressure to preserve foods. However, the principal mechanism of HPH is based on shear stress distribution in a material instead of a decrease in volume due to an increase in pressure as occurring in HPP. HPH can be used in extraction or preservation of bioactive compounds and phytochemicals. This review first describes the mechanism of HPH processing. Next, this review discusses the impact of HPH on extractability and stability of phytochemicals such as carotenoids, vitamin C, polyphenols, and anthocyanins in various food matrices. In general, the use of HPH slightly improved or maintained the extractability of the phytochemicals. Similarly, HPH slightly reduced or maintained the stability of the phytochemicals but this is dependent on the food matrix and type of phytochemical. HPH has a great potential to be used to improve the extractability and maintaining the stability of these phytochemicals or to be used together with milder thermal processing. Besides understanding the impact of HPH on the extractability and stability of phytochemicals, the impact of HPH on the nutritional quality of the food matrices needs to be thoroughly evaluated.

Keywords: anthocyanin, ascorbic acid, carotenoid, carotene, extraction, lycopene, phenolic acid, vitamin C

INTRODUCTION

Phyto is a Greek word that means “plant.” Hence, phytochemicals are chemicals derived from plants. They are secondary metabolites of plants that provide certain color, flavor, or for protection against pests and pathogens (Puri et al., 2012). Phytochemicals are small molecules in plant that constitutes about 10% or less of the plant matrix. They can be recovered from flowers, fruits, vegetables and herb using various extraction techniques (Harjo et al., 2004). Phenolic compounds belong to a large class of compounds with great diversity in their structures. Examples are simple phenolic acids that contain one hydroxyl group attached to an aromatic ring such as vanillin and caffeic acid. Besides simple phenols, there are polyphenols such as flavonoids and stilbenes in which the molecule bear a minimum of two phenolic rings. Example of these polyphenols include anthocyanin, proanthocyanins, and gallotannins (Cheynier, 2012). Carotenoid is a natural, lipid-soluble pigment occurring in plants, algae, and certain fungi. Carotenoid is located in the chloroplast and chromoplast of plants or bounded to a macromolecule such as fiber or protein (Serment-Moreno et al., 2017). Carotenoid contributes to yellow, orange and red color. The base structure of carotenoid consists of a branched five-carbon unit which is called isoprenoid (Boon et al., 2010).
Conventional extraction techniques for the recovery of phytochemicals can be carried out using solid-liquid extraction method. These include maceration, infusion, Soxhlet extraction, and steam distillation method (Brennan et al., 2013). Soxhlet extraction is conventionally used for recovery of phytochemicals from plants and is frequently used as the reference technique to evaluate other conventional and non-conventional extraction techniques (Wang and Weller, 2006). Conventional techniques usually employ the use of polar or non-polar organic solvents except for steam distillation and infusion technique that employ water as solvent (Haroen et al., 2013; Dhanani et al., 2017).

Extraction yield of conventional solvent extraction methods are usually high but the usage of solvents in these techniques limits the application of the extracted phytochemicals in foods as solvents may pose harmful effects when ingested (Selvamuthukumar and Shi, 2017). This issue may be solved by using food grade solvents but there are other challenges as well. These include long duration of extraction, high cost to use high purity solvent, and the use of heat treatment which will result in degradation of thermo labile phytochemicals (Azmir et al., 2013). Therefore, alternative approaches involving non-thermal or solvent-free extraction methods are highly desirable for application of phytochemicals in the food industry. Pulsed-electric field, supercritical fluid extraction and high-pressure processing are examples of non-conventional techniques (Azmir et al., 2013).

High pressure processing (HPP) applies pressure uniformly and transmit the pressure to a sample using a pressure transmitting medium at ambient or sub-ambient temperature for several minutes, without inducing a shearing effect. Although many food products such as fruit and vegetable beverages are commercially being treated with HPP for pasteurization, this process is not a continuous process and therefore only allowing a relatively low processing volume. On the other hand, high-pressure homogenization (HPH), also known as high-pressure valve homogenization or dynamic high-pressure homogenization, is an emerging continuous flow process technology that enables homogenization and pasteurization or in certain cases, sterilization of fluids in one single step (Levy et al., 2020). HPH can be used in the food industry to inactivate spoilage or pathogenic microorganisms, prepare emulsions, reduce particle size and improve the rheological properties of food products. Research on the use of HPH to extract or increase the functionality of bioactive compounds has been increasing since the year 2010. It was suggested that this trend is due to increasing interest in the food industry to improve nutritional and organoleptic quality of foods, besides concern for food waste valorization. HPH technology is a green technology as it does not require the use of polluting solvents, short processing times, low energy consumption and carbon dioxide emissions (Mesa et al., 2020). In this context, HPH technology has huge potential for applications in the food industry. One of the emerging applications is the use of HPH technology for extraction of phytochemicals. However, information on the application of HPH technology for extraction of phytochemicals has to be derived from individual studies conducted. Thus, the application of HPH technology for extraction of phytochemicals such as carotenoids and polyphenols, and its impact on their extractability is organized and discussed in this review. Phytochemicals are key components in foods that provide benefits to human health. As the main focus of most studies is either on the impact of HPH technology on microbial inactivation, physical stability or stability of macromolecules, this review also discusses the impact of HPH on the stability of various phytochemicals. The mechanism of HPH is also discussed.

### MECHANISM OF HPH

HPH is a homogenization technology that results in the alteration of the physical structures and microbial and enzyme inactivation of food products. Both HPH and HPP technology are able to achieve microbial and enzyme inactivation in food products. However, the application of HPP depends on the pressure whereas the application of HPH is dependent on the distribution of shear stress across a product (Augusto et al., 2018). The difference between HPH and HPP is summarized in Table 1. HPH functions at an elevated pressure of more than 100 MPa and is not limited to a single configuration. Figure 1 shows a general representation of HPH. With its homogenization pumps, valve and accessories, many configurations can be allowed but due to the particularities of the equipment such as valve design, only fluidic products and not particulate products can be used in HPH. Therefore, HPH process is considered as a continuous process (Augusto et al., 2018).

The main components of HPH consists of a homogenization valve and a high-pressure pump. A pressure of 34 MPa was considered as high-pressure homogenization in the early days; but 300 MPa or more can be achieved (Diels and Michiels, 2006). HPH was defined by Harte (2016) as homogenization processes whereby the pumps can reach at least 100 MPa to

| TABLE 1 | Summary of distinguishing high-pressure homogenization from high-pressure processing. |
|---------------------------------|---------------------------------|---------------------------------|
| Description | High-pressure homogenization (HPH) | High-pressure processing (HPP) |
| Principle | - Distribution of shear stress across products | - Equal distribution of pressure in all directions in products (isostatic principle and principle of Le Chateller) |
| | - Changes in the structure of the product | - Non-porous food will maintain their original shape |
| Pressure level | 100–350 MPa | 400–600 MPa |
| Temperature rise | 15–18◦C/100 MPa | 3–8◦C/100 MPa |
| | - Reversible upon decompression | - Reversible |
| State of product | Liquid | Liquid and solid |
| Mode of operation | Continuous | Batch and semi-continuous |
| Number of industrial applications | Few | Many |

Adapted from Martínez-Monteagudo et al. (2017) and Augusto et al. (2018).
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FIGURE 1 | Schematic representation of high-pressure homogenization technology.

a liquid food. The HPH process carried out at upper pressure range of 200 MPa or more is called an ultra-HPH (Marszałek et al., 2017). It should, however, be noted that the cut-off point between HPH and ultra-HPH can differ between authors. This is because HPH technologies are kept on evolving (Harte, 2016). Pressure is first intensified by the high-pressure pump that pressurizes a product that results in a fluid pressurization system. The high pressure generated can be up to 400 MPa and acts as the driving force that results in the flow of the fluid through and beyond the homogenization valve (Georget et al., 2014; Marszałek et al., 2017; Augusto et al., 2018). Although the high pressure is the driving force of the homogenizer, the main work occurs at the homogenization valve. The homogenization valve consists of a small orifice (in order of micrometers) in between the valve and the valve seat (Figure 2). At the orifice of the valve, a fluid is subjected to shearing action as its motion is disrupted by this orifice (Sanguansri and Augustin, 2006). Different phenomena such as cavitation, turbulence, collision and impingement contributed to these shearing effects (Martinez-Monteagudo et al., 2017). However, this review focuses on cavitation and turbulence.

SHEARING EFFECTS OF HPH

Cavitation
Due to abrupt reduction in pressure of the flowing liquid, cavitation occurs (Carlton, 2012) whereby cavities are formed followed by subsequent collapse within the liquid (Figure 3). This is a result of vaporized fluid condensation (Martinez-Monteagudo et al., 2017). At the orifice of the valve, a continuous decrease in pressure is occurring due to fluid acceleration. When the fluid vapor pressure is reached, this allows vaporization which is the transition of liquid into vapor. When the fluid leaves the valve, an increase in the flow area for the fluid decreases its velocity. Thus, the pressure at the exit increases back to atmospheric pressure. Due to this, a great amount of energy is released that causes great shear stress because of the condensation of vaporized fluid (Augusto et al., 2018).

Turbulence
Turbulence is a phenomenon that occurs when the motion of a fluid is suddenly reduced by about 100–1,000 times around the valve. As the area for flowing fluid is abruptly reduced, velocity increases according to the law of mass conservation. With increasing velocity around the valve and high magnitude of velocity gradients, there is non-uniform fluid motion that gives rise to turbulent flow (Martinez-Monteagudo et al., 2017). Turbulent flow (Figure 3) enhances diffusivity of mixing as well as improving mass and heat transfer. Turbulent flow also causes generation of heat through dissipation of kinetic energy (Rosa, 2006). The dissipative nature of turbulence is important as sufficient energy is required to breakup particle and cause the formation of droplets (Martinez-Monteagudo et al., 2017).

Heat of Homogenization
Even though HPH is considered as a non-thermal process, increase in the temperature of food materials is unavoidable because of adiabatic heating. Turbulence as well as shearing and mixing effects of liquid during homogenization produces adiabatic heating. Therefore, it is important to control and...
monitor the temperature of food materials after homogenization for prevention of over-heating and subsequent loss of heat-sensitive molecules (Diels and Michiels, 2006; Dumay et al., 2013). Heat exchanger (Figure 1) following the homogenization valve (Augusto et al., 2018) is generally used to control the temperature of HPH.

**IMPACT OF HPH ON EXTRACTABILITY OF PHYTOCHEMICALS**

Pre-treatment of phytochemicals with HPH before extraction is a relatively new approach. Hence, studies on the effect of HPH treatment on the extractability of phytochemicals is lesser studied in various food systems than studies on the stability of phytochemicals. Conventionally, phytochemicals are recovered from plant materials using solid-liquid extraction techniques but newer non-thermal and solvent-free approaches such as high-pressure extraction is being increasingly used for the extraction of phytochemicals (Cardoso et al., 2013; Casquete et al., 2014). The hypothesis is that the use of HPH process may increase the extractability of any compound such as phytochemicals (Martinez-Monteagudo et al., 2017). Table 2 summarizes various conditions used in HPH and the effects of HPH on phytochemicals in several food matrices. The model, valve type and scalability of the HPH used are mentioned as reported by the authors of these studies. Current knowledge on HPH valves in terms of design, flow rate, operating pressure and scalability has been discussed in the review by Martinez-Monteagudo et al. (2017).

**Phenolic Compounds**

There are limited studies on the impact of HPH on the extractability of phenolic compounds. Zhu et al. (2016) investigated the effect of HPH at a pressure of 158.58 MPa and alkaline treatment for the extraction of phenolic acids from potato peels. With HPH treatment, an increase of 27.4% in extraction yield was obtained compared to that without HPH treatment. Combination of HPH and alkaline treatment in the extraction of phenolic acids further improved the extraction yield.
by 44.4%. In these potato peels, the most abundant compounds are the free or bound phenolic acids (Samarin et al., 2012). HPH was suggested to cause structural changes to the cells by weakening the cell walls and releasing bound phenolic acids (Mattila and Kumpulainen, 2002).

HPH at 100 MPa for 1–10 passes with water as the process medium were used by Jurić et al. (2019) to extract valuable compounds from tomato peels. HPH treatment for 10 passes resulted in an increase in the extraction of polyphenols by 32.2% and was better than high-shear mixing at 5 min at 20,000 rpm. This was attributed to the release of intracellular compounds such as polyphenols is dependent mainly on the extent of cell disruption. These studies demonstrated positive effect of HPH on the extraction of phenolic acids from different food matrices, but they only utilized one pressure. Increasing the pressure of the HPH system or number of HPH cycles or passes is most likely going to increase the extractability of phenolic acids.

**Carotenoid**

The release of carotenoid by HPH processing is achieved through the mechanical disruption of plant cell walls. Plant cell walls are not able to withstand high shear stress occurring at the valve (Figure 3) of a high-pressure homogenizer (Colle et al., 2010a). Extraction of carotenoids from tomato peels into aqueous phase using HPH at 100 MPa for 1–10 passes was investigated by Jurić et al. (2019). Lycopene was found to be the main carotenoid in the pellet and supernatant of the aqueous phase after centrifugation. Interestingly, the pellet from aqueous phase treated by high-shear mixing at 5 min at 20,000 rpm contained higher amounts of lycopene than that by 5 HPH passes whereas the supernatant from the aqueous phase treated by 5 HPH passes contained higher amount of lycopene than that of high-shear mixing. It was suggested that lycopene can be extracted from the tomato peel structure into the aqueous phase using HPH. In addition, the yields of extraction of lycopene in this study was found to be comparable with other studies using solvent extraction assisted by cellulase and pectinase enzymes or ultrasound-assisted extraction but higher than those using pulsed electric field-assisted solvent extraction, supercritical carbon dioxide extraction, pressurized water extraction as well as conventional solvent extraction from dried tomato peels using various combinations of solvents (Jurić et al., 2019). Varying the
| Food Matrix      | Compound          | Equipment                                                                 | Conditions                                                                 | Effect                                                                 | References                      |
|------------------|-------------------|---------------------------------------------------------------------------|---------------------------------------------------------------------------|------------------------------------------------------------------------|---------------------------------|
| Tomato pulp      | Lycopene          | - Panda 2K, Gea Niro Soavi, Mechelen, Belgium                             | - Pressure: 8–132 MPa                                                   | 1. Lycopene content after HPH ranged from 87 to 102%                    | Cole et al., 2010a              |
| Mandarin juice   | Carotenoid        | - Panda 2 K, GEA Process Engineering Inc., Harlow, U.K.                   | - Pressure: 50–250 MPa                                                 | 1. HPH maintained carotenoid content                                   | Carreño et al., 2011            |
| Apple juice      | Polyphenol        | - Model/DRG No. FPG 11300:400 Hygienic Homogenizer, Stansted Fluid Power Ltd., Harlow, U.K. | - Pressure: 200 MPa                                                   | 1. HPH resulted in significant degradation of vitamin C                | Suárez-Jacobo et al., 2011      |
|                   | β-Carotene        | A ceramic valve able to withstand up to 400 MPa                          | - Flow rate: 270 mL/min                                                | 2. HPH resulted in loss of β-carotene                                  |                                 |
|                   | Vitamin C         | - Model FPG 7400H:350, Stansted Fluid Power Ltd., Essex, U.K.             | - Pressure: 50–250 MPa                                                 | 3. HPH maintained vitamin C content                                    |                                 |
| Fruit juices     | Vitamin C         | - A bench-scale high-pressure homogenizer (mm-GEN 7400 series by Stansted Power Fluids, UK) | - Flow rate: 0.7 L/h                                                  |                                                                       |                                 |
| Carrot puree      | β-Carotene        | - Panda 2K, Gea Niro Soavi, Mechelen, Belgium                             | - Pressure: 10 MPa                                                    | 1. Minimal isomerization of β-carotene was observed after HPH          | Lemmens et al., 2013            |
| Tomato puree      | Lycopene          | - Panda 2K, Gea Niro Soavi, Mechelen, Belgium                             | - Pressure: 30, 60, 90 and 130 MPa                                    | 1. HPH in inconsistent carotenoid content                              |                                 |
| Mango puree       | β-Carotene        | - Panda 2 K, Gea Niro Soavi, Mechelen, Belgium                             | - Pressure: 8–132 MPa                                                 | 1. HPH showed no significant influence on β-carotene content           | Knockaert et al., 2012a         |
| Tomato pulp      | Lutein            | - Panda 2K, Gea Niro Soavi, Mechelen, Belgium                             | - Flow rate: 20, 50 and 100 MPa                                       | 1. HPH resulted in loss of vitamin E                                   | Panozzo et al., 2013            |
| Orange juice     | Flavonoid         | - FPG 11300:400, Stansted Fluid Power Ltd., Essex, U.K.                   | - Pressure: 200 MPa                                                   |                                                                       | Velázquez-Estrada et al., 2013  |
|                   | Carotenoid        | A ceramic valve that was able to withstand up to 400 MPa                 | - Flow rate: 120 L/h                                                   | 1. HPH increased extractability of phytochemicals                      |                                 |
| Mulberry juice    | Anthocyanin       | - A bench-scale high-pressure homogenizer (UC-10C series by Guangzhou Juneng biology and technology Co., Ltd., Guangdong, China) | - Pressure: 200 MPa                                                   | 2. HPH significantly reduced anthocyanin content                       | Yu et al., 2014                 |
|                   | Vitamin C         | A high pressure disruption valve                                         | - Flow rate: 10 L/h                                                   | 3. HPH significantly reduced vitamin C content                         |                                 |
| Soya milk         | Polyamines        | - Model FPG11300, Stansted Fluid Powder Ltd., Essex, U.K.                 | - Pressure: 200, 300 MPa                                              | 1. HPH maintained the levels of polyamines                             | Toro-Funes et al., 2014a        |
|                   | Tocopherols       | Ceramic UHPH valve (Stansted Series FPG 9080)                             | - Tη: 55, 65, and 75°C                                               | 2. HPH resulted in 20–50% of losses in the tocopherol contents         |                                 |
|                   | Phytosterols      |                                                                       |                                                                       | 3. HPH increased the total phytosterol extractability                   | Toro-Funes et al., 2014b        |
| Almond beverage   | Polyamines        | - Model FPG11300, Stansted Fluid Powder Ltd., Essex, U.K.                 | - Pressure: 200, 300 MPa                                              | 1. HPH maintained the levels of polyamines                             |                                 |
|                   | Tocopherols       | Ceramic UHPH valve (Stansted Series FPG 9080)                             | - Tη: 55, 65, and 75°C                                               | 2. HPH resulted in 80–95% of losses in the tocopherol contents         |                                 |
|                   | Phytosterols      |                                                                       |                                                                       | 3. HPH increased the total phytosterol extractability                   |                                 |
| Bovine milk       | Vitamin C         | - Stansted Benchtop HPH nG12500, Stansted Fluid Power Ltd., Essex, UK    | - Pressure: 300 MPa                                                   | 1. HPH retained 81% of vitamin C                                      | Amador-Espejo et al., 2015      |
|                   | α-tocopherol      | - A ceramic valve that was able to support 350 MPa                       | - Flow rate: 8 L/h                                                   | 2. HPH maintained the level of α-tocopherol                            |                                 |

(Continued)
TABLE 2 | Continued

| Food Matrix | Compound | Equipment | Conditions | Effect | References |
|-------------|----------|-----------|------------|--------|------------|
| Strawberry juice | Polyphenol | Nano Dispenser—NLM 100, South Korea | Pressure: 60 and 180 MPa, Passes: 2 and 5 | 1. HPH maintained total phenolic content at 60 MPa but increased total phenolic content at 180 MPa | Karacem et al., 2015 |
| Almond milk and B2 | Vitamins B1 | MRI in-house HPH system | Pressure: 350 MPa, Flow rate: 0.5 mL/s, Tm: 85°C | 1. HPH maintained the levels of vitamins B1 and B2 | Briviba et al., 2016 |
| Potato peel | Phenolic acid | Brand name was not provided by authors | Pressure: 158.8 MPa, Cycles: 2 | 1. HPH improved the extraction of phenolic acids | Zhu et al., 2016 |
| Apple juice | Polyphenol | Nano homogenize machine, ATS Engineering Inc., Canada | Pressure: 250 MPa | 1. HPH decreased the total phenolic content of apple juice | He et al., 2016 |
| Grape juice Orange juice | Polyphenol | - Two-stage HPH, SPX Flow Technology, Stansted Fluid Power Ltd., Essex, UK | Pressure: 10–100 MPa, Cycles: 1 and 10 | 1. HPH increased the extraction yield of vitamin C by 152.8% for floret and 44.1% for stalk | Yang et al., 2018 |
| Tomato juice | Carotenoids | A pilot-scale HPH (FGP11300:350, Stansted Fluid Power Ltd., Essex, UK) | Pressure: 246 MPa, Flow rate: 90 L/h, Tm: 99°C | 1. HPH increased β-carotene and total lycopene content | Yan et al., 2017 |
| Broccoli | Vitamin C | HOMOLAB, FBF, Italy | Pressure: 10–100 MPa | 1. HPH increased the extraction of lycopene | Jurić et al., 2019 |
| Tomato peels | Lycopene | An orifice valve assembly (orifice diameter of 150 µm) | Pressure: 100 MPa, Tm: Below 24°C, Passes: 10 | 2. HPH increased the total phenolic content of grape and orange juices | Karacem et al., 2015 |
| Carrot juice | Carotenoids | JN02HC, Guangzhou Juneng, China | Pressure: 20, 60, and 180 MPa, Tm: 25, 50, and 70°C, Passes: 1, 2, and 3 | 1. HPH increased total carotenoid bioaccessibility | Liu et al., 2019c |
| Carrot beverage | Carotenoids | JN-02HC, Guangzhou Juneng, China | Pressure: 60, 120, and 180 MPa, Tm: 25 and 60°C, Passes: 1 and 3 | 1. HPH resulted in better total carotenoid content during storage | Liu et al., 2019b |
| Carrot juice | Carotenoids | JN02HC, Guangzhou Juneng, China | Pressure: 20, 60, 100, 150, and 180 MPa, Tm: 25, 50, and 70°C, Passes: 1, 2, and 3 | 1. HPH maintained total carotenoid content | Liu et al., 2019ca |
| Strawberry nectar | Anthocyanin | PFK FPG12800, Stansted Fluid Power, U.K. | Pressure: 50, 100, 150, and 200 MPa, Tm: 25°C, Passes: 1, 3, and 5 | 1. HPH slightly affected anthocyanin content | Moscovici Joubran et al., 2019 |
| Kiwi juice | Polyphenol | PANDA (GEA, Parma, Italy) A R-type valve | Pressure: 200 MPa, Tm: 4°C, Cycles: 2 and 3 | 1. HPH increased total phenolic content | Patrignani et al., 2019 |
| Rosehip nectar | Carotenoids | Two-stage HPH (GEA Niro Soavi-Panda Plus 2000 Homogenizer, Parma, Italy) | Pressure: 75, 100, and 125 MPa, Cycles: 1 and 10 | 1. HPH increased total carotenoid content with increasing passes | Saricaoglu et al., 2019 |
| Lettuce waste | Polyphenol | A continuous lab-scale high-pressure homogenizer (Panda Plus 2000, GEA Niro Soavi, Parma, Italy) | Pressure: 80 and 150 MPa, Flow rate: 10 L/h, Cycles: 1 and 10 | 1. HPH partially maintained phenol content | Plazzotta and Manzocco, 2019 |
| Peach juice | Ascorbic acid | Two-stage HPH, SPX Flow Technology, soeborg, Denmark | Pressure: ~55 MPa, Tm: 45°C | 1. HPH increased ascorbic acid content | Yıldız, 2019 |
| Pomelo and kiwi juices | Polyphenol | ATS Engineering Inc., Canada | Pressure: 250 MPa | 1. HPH increased total phenolic content | Quan et al., 2020 |
| Mixed carrot, apple, and peach juices | Polyphenol | JN-02HC series, Guangzhou Juneng, China | Pressure: 140 MPa, Tm: 25°C | 1. HPH increased total phenolic content | Weilala et al., 2020 |
pressure of the HPH system or number of HPH cycles or passes is most likely going to affect the extractability of carotenoids.

**IMPACT OF HIGH-PRESSURE HOMOGENIZATION ON THE STABILITY OF PHYTOCHEMICALS**

HPH process is conventionally utilized mainly for enzyme and microbial inactivation in the food industry (Welti-Chanes et al., 2009; Carreño et al., 2011). The decrease of microbial count in HPH-treated food material is usually not accompanied by a loss in nutritional value (Hayes et al., 2005; Calligaris et al., 2012). As HPH is a non-thermal process, it is usually regarded that HPH would cause minimal impact on the stability of phytochemicals. The following sections discussed the impact of HPH on the stabilities of vitamin C, phenolic compounds and carotenoid in several food matrices.

**Vitamin C**

The stability of L-ascorbic acid is influenced by the presence of light, oxygen and temperature (Volf et al., 2014). The retention of L-ascorbic acid in mango nectar with combination of HPH at 200 MPa and heat treatment (61.5 and 75.5°C) was investigated by Tribst et al. (2011). In addition, HPH treatments without combination with heat treatment at 200 MPa and 300 MPa were also investigated. The retention of L-ascorbic acid in mango nectar was found to be low. HPH treatment resulted in ~50% losses of L-ascorbic acid. However, this is in contrast with the study of Pérez-Conesa et al. (2009), who reported that HPH at 10, 15 and 20 MPa did not reduce the L-ascorbic acid content in tomato. This was most likely due to the low pressure that was employed in the HPH treatment (Zhou et al., 2017). Besides, the presence of oxygen in the mango nectar and increase in temperature during HPH could result in degradation of L-ascorbic acid (Tribst et al., 2011). Hence, deaeration to remove oxygen is recommended prior to HPH processing.

The effect HPH treatment from 50 to 250 MPa on commercial orange, red orange, pineapple fruit juices and Annurca apple juice was investigated (Maresca et al., 2011). The vitamin C concentration of the HPH treated juice was reported to be similar to the fresh juices. It was suggested that the HPH treatment did not significantly cause the degradation of vitamin C in the juices (Maresca et al., 2011). The effect of HPH at 100, 200, and 300 MPa on L-ascorbic acid content in orange juice in comparison to thermal pasteurization (90°C for 1 min) was investigated by Velázquez-Estrada et al. (2013). There was a gradual decrease of L-ascorbic acid content (1.7, 4.6, and 10.7%) at pressures of 100, 200, and 300 MPa, respectively. However, HPH-treated juices retained L-ascorbic acid better than pasteurized juice with a decrease of 20.1%. This study is in accordance with another study using apple juice (Suárez-Jacobo et al., 2011). In contrast, the degradation of L-ascorbic acid due to a gradual increase in HPH pressure is not in accordance with the study by Welti-Chanes et al. (2009) in which the stability of L-ascorbic acid in orange juice remained stable after HPH treatment at 50–250 MPa. The degradation of L-ascorbic acid during storage is deduced to be more influenced by oxygen, temperature and light instead of the initial HPH treatment (Sharabi et al., 2018). The presence of trace metals in the processing equipment could also promote the degradation of L-ascorbic acid (Ball, 2006; Tribst et al., 2011).

The effect of HPH treatment at 75, 100, and 125 MPa on the ascorbic acid content of rosehip nectar was investigated by Sarıaçoğlu et al. (2019). HPH treatment was found to decrease the ascorbic acid content of rosehip nectar as compared to the control. There were no significant differences between the ascorbic acid content of rosehip nectar treated at 75 and 100 MPa but the lowest ascorbic acid content in rosehip nectar was obtained using 125 MPa (Sarıaçoğlu et al., 2019). HPH treatment at approximately 55 MPa for 3 min with an inlet temperature of 45°C of peach juice was found to be better than heat treatment at 72°C for 15 s in retention of ascorbic acid and total phenolic contents. However, ultrasonic homogenization at a frequency of 20 kHz was better than HPH treatment in retention of ascorbic acid and phenolic compounds of the peach juice (Yildiz, 2019). Overall, ascorbic acid is susceptible to degradation during HPH treatment.

**Phenolic Compounds**

The stability of phenolic compounds after isolation from their respective plant matrix is crucial as they are susceptible toward degradation in the presence of thermal stress or oxygen from the external environment (Karaaslan et al., 2013). The stability of anthocyanin in bilberry juice upon HPH and thermal treatment was investigated (Frank et al., 2012). Encapsulation of bilberry juice in monolayers of triglycerides was first carried out followed by thermal treatment, HPH at pressures from 30 to 150 MPa or combination of both thermal and HPH treatment. The anthocyanin content in the bilberry juice after HPH treatment remained relatively similar to untreated juice but thermal treatment caused degradation of the anthocyanin content. This study shows that the stability of anthocyanin is maintained even after mechanical stress induced by HPH treatment (Patras et al., 2010; Marszałek et al., 2017).

The impact of HPH on the polyphenol content of apple juice was investigated (Suárez-Jacobo et al., 2011). Fresh apple juices were subjected to HPH treatment at pressures of 100, 200, and 300 MPa or thermal treatment at 90°C for 4 min. The total polyphenol content of apple juices decreased (10.6, 6.0, and 1.4%) when subjected to HPH treatment under pressure of 100, 200, and 300 MPa, respectively, as compared to pasteurized juices. This indicates that the polyphenols were more stable with increasing pressure. Nevertheless, polyphenols in apple juice are generally not stable due to the presence of polyphenol oxidase, an enzyme that can cause degradation of polyphenols (Buckow et al., 2009). It was suggested that polyphenol oxidase remained stable at low pressures lower than 200 MPa and oxidized the polyphenols. However, the high pressure of 300 MPa must have inactivate the enzyme (Schilling et al., 2008), hence, there was no significant impact of HPH on total polyphenol content of apple juice. On the contrary, 300 MPa was reported in other studies to unable to inactivate PPO in apple juices (Saldo et al., 2009; Mckay et al., 2011).
Velázquez-Estrada et al. (2013) investigated the effect of ultra HPH and thermal pasteurization on the properties of orange juice. Orange juice was heated at 90°C for 1 min for thermal pasteurization whereas for HPH processing, orange juice was pressurized to 100, 200, and 300 MPa. The polyphenol content decreased by 0.77% (100 MPa), 1.54% (200 MPa), and 6.61% (300 MPa). Although polyphenol content was reduced because of HPH treatment, it was significantly lower than that of thermal pasteurization which showed a decrease of 19.0%. This study showed that the degradation of polyphenols was minimized under non-thermal treatments (Oancea et al., 2018).

There was no effect of HPH treatment at 60 MPa on the total phenolic content of strawberry juice. However, HPH treatment at 100 MPa resulted in an increase on the total phenolic content of strawberry juice. Increasing HPH passes from 2 to 5 also resulted in an increase on the total phenolic content of strawberry juice (Karacam et al., 2015). HPH treatment at 250 MPa for 10 min decreased the total phenolic content of apple juice but increased the total phenolic contents in grape and orange juices. It was suggested that HPH treatment ruptured the cellular structure of grape and orange fruits, favoring the release of bound phenolic substances from these fruit cells and hence enhancing their phenolic content but for apple juice, size reduction of plant tissues might have led to the interaction of cytoplasmic polyphenol oxidase and phenolic compounds in the vacuoles and result in oxidative degradation of the phenolic compounds (He et al., 2016).

The effect of HPH processing (200 MPa and thermal pasteurization (95°C for 1 min) on the polyphenol content of mulberry juice was investigated by Yu et al. (2014). HPH caused a reduction in the anthocyanin content, cyanidin 3-rutinoside and cyanidin-3-glucoside by 33.2 and 38.8%, respectively, in mulberry juice. This was due to the presence of PPO. This study is in accordance with the study on apple juice in which a pressure of 200 MPa was found to be insufficient to inactivate PPO (Suárez-Jacobo et al., 2011). HPH processing may promote the oxidation of PPO through the exposure of active sites as a result of a change in its conformation (Liu et al., 2009; Bot et al., 2018). However, the pressure used in HPH processing has to be high enough (>300 MPa) to result in inactivation of PPO.

The inactivation of oxidative enzymes by the use of blanching prior to HPH produces a color-stable juice but blanching can cause significant reduction in phenolic content. This is because the applied heat and leaching effect in the water used in blanching. Ground lettuce waste was firstly homogenized at 40 MPa before proceeding to HPH treatments at 80 and 150 MPa. The combination of HPH with a blanching pre-treatment resulted in a homogeneous lettuce juice with partially maintained polyphenolic content (Plazzotta and Manzocco, 2019). Rosehip nectar treated at 125 MPa resulted in higher total phenolic content than that treated at 75 MPa. This was attributed to the decreasing particle size and releasing of materials from the cells (Saricaoglu et al., 2019). The total phenolic content of kiwifruit juices with ultra HPH treatment at 200 MPa for 3 cycles significantly increased in comparison to the control from 35 to 42 mg/100 mL of juice. These results indicate that an increase in the availability of phenolic compounds (Patrignani et al., 2019).

The effect of HPH treatment up to 200 MPa on the total phenolic content of strawberry nectar was investigated by Moscovici Joubran et al. (2019). The anthocyanin content was only slightly affected by the HPH pressure levels. The total phenolic content was not affected by HPH pressure levels but significantly increased up to 30% following number of HPH passes. This was attributed to more extraction of polyphenols from the achenes and pulp of strawberry (Moscovici Joubran et al., 2019). HPH treatment at 140 MPa, 25°C was found to enhance the total phenolic content of carrot, apple and peach mixed juices. This was attributed to the release of more polyphenols from the vacuoles of different fruits as a result of intensive cell disruption (Wellala et al., 2020). There were no significant differences between HPH-treated (250 MPa for 10 min) and thermally treated (80°C for 30 min and 90°C for 30 s) kiwi and pomelo juice in terms of total phenolic content. Both treatments showed an increase of 10.6–17.5% in total phenolic acids. This was attributed to both treatments were able to damage plant cell walls that assisted in the release of phenolic compounds (Quan et al., 2020). Overall, HPH processing affects the stability of phenolic compounds in juices but depending on the pressure that was applied and number of HPH passes.

Carotenoid

Carotenoid content in fruits is depending on their stage of developmental and environmental growth conditions. Lutein is the most representative carotenoid at the green stage of fruit, which reflects the characteristic of chloroplastic tissues, followed by β-carotene, violaxanthin and neoxanthin (Choo, 2019). Carotenoids contain a conjugated system of double bonds that are susceptible to oxidation and isomerization (Boon et al., 2010). These chemical reactions occur frequently during thermal processing due to high heat and exposure to oxygen. HPH processing may maintain the stability of carotenoid as HPH is non-thermal technology.

The effect of HPH has been investigated in tomatoes, carrots and mango fruits. It was found that the effect of HPH process on tomatoes varied according to the homogenization pressure (8.4–132.7 MPa). The total lycopene content ranged between 87 and 102% following these different pressure treatment (Collé et al., 2010a). This variation in the lycopene content may be due to the variability between the tomatoes as well as ripeness of the tomatoes (Martínez-Valverde et al., 2002). Knockaert et al. (2012a) compared the equivalent thermal and high pressure sterilization processes of HPH-treated tomato puree with olive oil. No significant differences were found between the thermal and high pressure sterilization processes. Panozzo et al. (2013) investigated the effect of HPH at pressures of 20, 50, and 100 MPa on carotenoid content in red, orange and yellow tomato cultivars. The type of carotenoid in each tomato differed according to the types of tomatoes. Lycopene is the predominant carotenoid in red tomatoes whereas lutein was the predominant carotenoid in yellow tomatoes. Carotene was found predominantly in orange tomatoes (Rizk et al., 2014; Chaudhary et al., 2018). This variability in the type of predominant carotenoid was also due to the ripeness of the tomatoes (Zeng et al., 2015). Lutein in the globular chromoplasts...
was found to be more resistant to rupture than the lycopene-residing crystallloid ones. The crystalline state of lycopene is more fragile and can be easily ruptured by mechanical stress (Schweiggert et al., 2012). Panozzo et al. (2013) found that tomato cells were completely disrupted with HPH treatment at 20 MPa which ease the release of cellular content including lycopene, lutein and carotene. The increase in homogenization pressure further disrupted the cell material with only fragments of cells were observed for HPH treatment at 100 MPa (Panozzo et al., 2013).

The effect HPH treatment on the isomerization and degradation of lycopene in tomato puree was investigated by Knockaert et al. (2012b). The lycopene content of HPH treated tomato puree was similar to the untreated puree. It was suggested that HPH at 10 MPa was insufficient to result in the disruption of the cell walls of tomato puree (Knockaert et al., 2012b; Kubo et al., 2013; Zhou et al., 2017). A pressure >50 MPa was recommended. In addition, the lycopene in HPH treated tomato puree remained as the stable trans-lycopene at 85.4% and cis-lycopene was at 14.6%. There was no formation of cis-lycopene in untreated tomato puree. In tomato pulp, intense thermal pasteurization caused elevation of cis-lycopene from 14.6 to 26.8% (Colle et al., 2010b). Furthermore, 13-cis-lycopene was the predominant degradation product of lycopene (Khoo et al., 2011; Lemmens et al., 2013).

The stability of carotene in mango puree upon HPH and thermal treatment (80–150°C for 20 min) was investigated by Shi et al. (1999). At 30, 60, 90, and 130 MPa, the ratio of β-carotene/total β-carotene was found to be 0.65, 0.63, 0.62, and 0.55, respectively. Isomerization of the stable trans-β-carotene was minimal as well. Isomerization of trans-β-carotene into cis-β-carotene would reduce its bioactivity and increase the susceptibility toward oxidation (Ball, 2006). The impact of HPH on the degradation and isomerization of carotenoids is depending on the pressure that was applied during HPH processing.

Carreño et al. (2011) studied the effect of HPH at pressure range of 0–120 MPa on the total carotenoid content of mandarin juice as compared to thermal pasteurization (90°C for 10 s) or HPP at 0–450 MPa. There was a decrease in total carotenoid content observed using HPH and HPP processing at 1.66 and 5.54%, respectively. However, thermal processing was shown to increase the total carotenoid content by 6.80%. Increasing pressure resulted in greater loss of carotenoid although all three treatments did not cause significant decrease of carotenoid content. The release of carotenoid from HPH processing is due to the mechanical disruption of plant cells within the chromoplast (Colle et al., 2010a; Palmero et al., 2016a,b).

Yan et al. (2017) investigated the impact of two high pressure techniques (HPH treatment at 246 MPa, 99°C, <1 s and HPP at 600 MPa, 46°C, 5 min) and compared to that of thermal processing at 90°C, 90 s of tomato juice. Total lycopene content and isomerization in the tomato juice were not significantly affected by the HPH and HPP treatment. Similar results were obtained for β-carotene content. Thermal processing resulted in significant decrease in β-carotene content of tomato juice. It was proposed that these two pressure-based technologies have the potential for application in tomato juice processing with good carotenoid retention (Yan et al., 2017).

HPH treatment assisted by moderate inlet temperature (180 MPa, 1 pass and 60°C) showed better total carotenoid content of a carrot beverage stored at 4°C for 28 days than the combination of HPH and heat treatment at 90°C for 5 min (Liu et al., 2019b). Rosehip nectar treated at 125 MPa resulted in higher total carotenoid content than that treated at 75 MPa. In addition, the total carotenoid content increased with increasing HPH passes from 1 to 3 passes. Lycopene is the main carotenoid in rosehip fruit (Saricaoglu et al., 2019). The effect of HPH treatment of carrot juice at 20, 60, 100, 150, and 180 MPa on the total carotenoid content of carrot juice was investigated by Liu et al. (2019a). The total carotenoid content was unaffected by the HPH treatment. HPH treatment at 180 MPa with moderate inlet temperature of 60°C showed a better preservation of carotenoids in carrot beverage than HPH treatment combined with heat treatment at 90°C for 5 min. During storage of the carrot beverage, the half-life, t1/2 of carotenoids (24.60–40.29 day) by HPH treatment was higher than the value of 19.21 day by heat treatment. This signified the negative effect of heat treatment on carotenoid retention during storage (Liu et al., 2019c). Overall, carotenoids remain quite stable during HPH treatment.

**CONCLUSIONS**

HPH is a promising technology that could open up new opportunities to preserve phytochemicals of liquid food products. Studies on using HPH have shown relative improvement in recovery of carotenoids and phenolic compounds from plant matrix. The impact of HPH treatment on the stability of phytochemicals is dependent on the type of phytochemical. Generally, the stability of carotenoids in various food matrices upon HPH treatment is maintained. The stability of polyphenols upon HPH treatment is dependent on the HPH pressure that was applied and number of HPH passes. Conversely, the stability of vitamin C is affected by HPH treatment. The HPH pressure applied, number of HPH cycles or passes and inlet temperature of a HPH system are key factors that will have an impact on the extractability and stability of phytochemicals. In addition, the HPH valve type or design and/or scale of HPH may also be the factors that can have an impact on the extractability and stability of phytochemicals. More studies to investigate these factors should be carried out. Nevertheless, HPH has a great potential to be used as treatment to improve the extractability and maintaining the stability of phytochemicals. HPH can also be used together with milder thermal processing. More research on the impact of HPH on the nutritional quality of food products should also be investigated in the future.
AUTHOR CONTRIBUTIONS

SY: original draft preparation and writing. CS: review. WC: conceptualization, writing, review, and editing. All authors contributed to the article and approved the submitted version.

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