Stabilization of wines with polymers and new bio-based carbon materials.

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Abstract. Wine is a complex product which changes its properties at every production stage, however due to the different processes which take place in the production stage can result into the formation of unwanted turbidity, deposition or can lead to distortion of taste. Despite the advances in improving wine stabilization processes, the search for new materials continues. The present work focuses on clarification of wines on the basis of new polymers and carbon materials obtained from bio-renewable raw materials and byproducts from the production of 2,5-hydroxymethylfurfural (5-HMF).

1 Introduction

An important quality for all types of wines - red, pink and white - is the preservation of their unique qualities, taste, color and aroma. Oxidative processes inherent in such complex systems as wine, lead to deterioration in taste, loss of brightness and formation of oxidation tones, which in turn reduces the quality of wine. To solve this problem, various technological methods are used to reduce oxidation at all the technological stages, by selecting suitable temperature regimes, as well as selecting various clarifiers and stabilizers. The wine clarification process involves removing suspended and colloidal particles that create turbidity, as well as removing unstable proteins or other macromolecules that can later denature or aggregate, resulting in a cloudy appearance of the wine after bottling. The purpose of stabilization is to ensure the clarity of the wine during aging and storage. Stabilization is carried out by removal (adsorption) of polyphenolic, protein and crystalline compounds with gluing agents, as well as filtration using membranes [1-4]. Traditionally, these are bentonite of various modifications, fish glue, albumin, chitosan (and others) and polymer compounds – PVP, PAA, and POE in combinations required for a specific stabilization method [5].

In addition, the removal of metal ions (primarily Fe (II), Cu (II)), Ni (II), as well as sulfur compounds that can get into the wine material as a result of processing grapes from various diseases is of great importance in the stabilization and safety of wines, from soil and technological operations [6-9].

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By developing and studying new polymeric materials and adsorbents, primarily based on bio-renewable plant raw materials, we can get an idea about the possibility of their targeted use to obtain high-quality wines.

The aim of the current research was to study the brightening and stabilizing properties of new copolymers based on 2,5-diallylfuranate and acrylic (AAc/DAF) or methacrylic acids (MAAc/DAF) which were obtained from bio-renewable raw materials.

2 Materials and methods

The efficiency of removing metal cations and polyphenolic compounds was evaluated using two model systems from Rkatsiteli and Cabernet-Sauvignon wines. The mixture of dry white wine of the Rkatsiteli variety, dry red wine of the Cabernet-Sauvignon variety was selected with the addition of tannin (mass concentration 80 and 1000 mg/dm³, respectively) and metal salts Cu (II) and Ni (II) (mass concentration 20 mg/dm³, respectively). The amount of introduced substances was determined based on the recommended safe values and analysis of literature data [10, 11].

Physicochemical indicators of wines were determined according to standardized methods operating in the territory of the Russian Federation; dry wines of the varieties Rkatsiteli, Cabernet-Sauvignon, obtained according to the schemes for the preparation of white and red dry wines. The investigated model wine samples were treated with various sorbents to establish optimal dosages and achieve stability.

As sorbents, synthesized polymeric materials from bio-renewable raw materials [12] were used together with activated carbons obtained from wastes of the agricultural industry and the production of 5-HMF [13]. The total exchange capacity for binary copolymers DAF, depending on the amount of the cross-linking agent (2.5–10 mol%), was 8.3–9.5 meq/g, and the sorption capacity for heavy metal ions was 365 mg/g and 290 mg/g (dry adsorbent) respectively for Ni (II) and Cu (II). The specific surface area of activated carbons according to the BET method was 600–860 m²/g (coal obtained by carbonization of humines - up to 860 m²/g; coal obtained by carbonization of beet pulp - 600 m²/g). For comparison, bentonite (in the form of a 10% suspension) was used. The studied dosages of sorbents were 2.0 g/dm³.

The brightening ability of the studied polymers and CMs was determined by the colorimetric method. The content of phenolic substances was determined according to the method recommended by the International Organization of Wine and Viticulture (OIV).

The efficiency of removing heavy metal ions was determined by analyzing the difference in the concentration of these metals in wine before and after adsorption. The concentration of heavy metals was determined by the method described in [12].

All determinations and tests were carried out in triplicate (repeated three times).

3 Results and discussion

Bentonite is used in the wine industry to remove proteins that are a potential source of cloudiness in wine. Through interaction with positively charged hydrocolloids and adsorption, bentonite interacts not only with proteins, but also with other molecules. Bentonite is used in combination with other adsorbents such as activated carbons or various polymers (polyvinylpolypyrrolidone, polyacrylamide, etc.). As a control, we used Plaxbenton bentonite (Table).

Copper at a concentration of 5 mg/dm³ negatively affects the taste of wine. Ferrous iron does not cause turbidity, but, being oxidized by atmospheric oxygen to Fe (III), combines with phenolic substances to form chelate compounds. As a result, sparingly soluble metallotanide-protein complexes are obtained; causing clouding of wines and the appearance
of sediment. The complex bound iron is not adsorbed by bentonites. In this case, mixtures of
tenonite with other adsorbents or complexing agents are used [14, 15].

It is on the binding of metals that the action of the AAc/DAF and MAAc/DAF copolymers
is directed, which also show activity against pathogenic micro flora [12]. After the treatment
of wine samples with an initial content of Ni (II) and Cu (II) of 20 mg/dm³ each, with the
copolymers the content of heavy metal ions decreased almost 10 times and was less than
3 mg/dm³.

Activated carbons with a high specific surface area and high porosity are widely used as
sorbents. Joint stepwise introduction of sorbents (bentonite + CM2, polymers + CM2) showed the maximum removal of tannins, with the formation of clumping, easily separable
sediments.

The carbon materials considered in this study were obtained from humic or beet pulp
using one-step and two-step processes involving chemical and physical activation methods
(denoted as Chem and Phys, respectively). The preparation conditions for each carbon
material (temperature and amount of modifier) are reflected in the names of the samples. For
example, Chem - 900 - 4 refers to a carbon material obtained by chemical activation of
humins at 900 °C and a mass ratio of KOH/humins of 4:1; Chem * - 900 - 1.5 refers to a
carbon material obtained by chemical activation of beet pulp at 900 °C and a mass ratio of
H₃PO₄/pulp of 1.5:1. A detailed description of the preparation methods and characteristics of
the obtained materials are given in [13]. Samples of CM from beet pulp were obtained in a
similar way. For the tests, materials with a maximum specific surface area of 600–860 m²/g
were selected; the results of processing model wine solutions are presented in Table.

**Table.** Chromatic characteristics of model wine solutions, where CA – activated carbon, CM1– beet
pulp carbon material, CM2 – carbon material from production waste 2,5 -HMF

| №   | Concentration of sorbents, by 2,0 g/dm³ | Model white wine solution | Model red wine solution |
|-----|--------------------------------------|---------------------------|------------------------|
|     |                                       | Optical characteristics   | Optical characteristics |
|     |                                       | Brightness, Y, %       | Chromaticity, λ, nm    | Clarity, P, %       | Brightness, Y, %       | Chromaticity, λ, nm    | Clarity, P, %       |
| 1   | Control (without additives)          | 88,40                    | 589,0                  | 8,0                   | 29,21                  | 540,0                  | 47,0                   |
| 2   | Bentonite                           | 87,10                    | 589,0                  | 8,4                   | 24,65                  | 538,6                  | 46,5                   |
| 3   | CA (control)                        | 86,90                    | 580,4                  | 8,1                   | 26,72                  | 539,4                  | 46,9                   |
| 4   | CM 1*                               | 85,88                    | 589,1                  | 7,5                   | 28,11                  | 539,7                  | 45,2                   |
| 5   | CM 2**                              | 82,08                    | 588,2                  | 8,7                   | 25,19                  | 538,5                  | 46,9                   |
| 6   | Bentonite + YA                      | 83,20                    | 586,1                  | 8,5                   | 23,01                  | 539,0                  | 47,2                   |
| 7   | Bentonite + CM 1*                   | 81,15                    | 584,8                  | 8,9                   | 22,48                  | 539,0                  | 47,9                   |
| 8   | Bentonite + CM 2**                  | 81,00                    | 581,1                  | 9,9                   | 21,11                  | 538,2                  | 50,2                   |
| 9   | Polymer AAc/DAF                      | 81,06                    | 585,4                  | 5,0                   | 25,14                  | 539,1                  | 49,1                   |
| 10  | Polymer AAc/DAF + YA                | 76,70                    | 584,2                  | 5,5                   | 24,43                  | 539,1                  | 49,4                   |
| 11  | Polymer AAc/DAF + CM 1              | 76,00                    | 574,0                  | 8,1                   | 23,15                  | 538,1                  | 51,8                   |
| 12  | Polymer AAc/DAF + CM 2              | 71,30                    | 570,1                  | 10,0                  | 22,15                  | 537,0                  | 52,4                   |

* – beet pulp carbon material Chem* -900- 1,5;
** – Carbonaceous material from humins Chem-900-4;

From the table it can be seen that upon the treatment of the sample solutions with the
studied sorbents leads to an increase in brightness and purity, and the combined pasting with
the polymers based on bio-renewable raw materials with new carbon materials makes it
possible to achieve brightness and brilliance In addition, the removal of metal ions that are
naturally present in wine (such as iron, nickel and copper) and catalyze the oxidation of
volatile aromatics, which cause darkening and cloudiness of the wine, contributes to more stable wines.

4 Conclusion

The use of AAc/DAF and MAAc/DAF copolymers for the treatment of model wines - promotes the selective removal of heavy metals and partial removal of phenolic compounds, which prevents browning and precipitation.

Currently, interest in the problem of obtaining wines free from heavy metals and other pollutants using polymeric materials remains relevant. Therefore, work is underway to create new polymer materials, which are medium- or weakly cross-linked polymer networks with hydrophilic properties, can be used as traditional cation exchangers, adsorbents, depending on the degree of cross-linking and the ability to swell and absorb metal ions. In addition, carbon materials obtained from humic waste from 5-HMF production showed good results in stabilizing model wines, which are not inferior to control carbon materials. At the same time, the processing of agricultural waste or 5-HMF synthesis products will make it possible to switch to bio-renewable raw materials and will help reduce the carbon footprint. The obtained materials showed good results on the sample solutions, which confirmed the possibility of their use for the stabilization of wines. To develop practical recommendations for their use in winemaking, it is necessary to conduct a more thorough comprehensive research in wines.

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