1. Introduction

Ceramics based on silicon nitride are well-known as low density materials with high strength and toughness. With this combination of properties silicon nitride based ceramics are an ideal candidate for several structural applications, even at high temperatures. Lately, by in-situ tailoring the microstructure, new observations were performed on structural and morphological development on silicon nitride ceramics in order to understand the governing principles of sintering processes [1, 2]. In this way, through formation of tough interlocking microstructure mechanical properties may be improved. Intensive research work has been done to improve physical and mechanical properties of silicon nitride ceramics through nanocomposite processing [3]. To increase the fracture toughness, incorporation of various energy-dissipating components into ceramic matrices have been performed [4, 5].

These components can be introduced in whisker, platelets, particles or fibre forms. A low cost silicon carbide-silicon nitride nanocomposite processing route has been reported by Hnatko et al [6]. In this case, the formation of bulk silicon nitride based nanocomposite is realized by carbothermal reduction of SiO₂ by carbon in the Y₂O₃–SiO₂ system at the sintering temperature, with SiC nanoparticles as result. Although, the mechanical properties of as-prepared samples should be further optimized, this process seems to be a perspective choice for silicon nitride-silicon carbide nanocomposite production [7]. In this paper silicon nitride based nanocomposites were prepared through carbon black addition by mechanochemical synthesis and hot isostatic pressing (HIP). Results about structural, morphological and compositional measurements are presented.

2. Experimental method

Details about sample preparation can be followed in Table 1. The compositions of the starting powder mixtures of the four materials were the same: 90 wt.% Si₃N₄ (Ube SN-ESP), 4 wt.% Al₂O₃, and 6 wt.% Y₂O₃. In addition to batches carbon black (Taurus Carbon black, N330, average particle size between ~50–100 nm), graphite (Aldrich, synthetic, average particle size 1–2 μm) and carbon fibre (Zoltek, PX30FBSW08) were added. The powder mixtures were milled in ethanol in a planetary type alumina ball mill for 150 hours. The samples were compacted by dry pressing at 220 MPa. Samples from 644 and 645 batches were collected during milling process after several stops. Carbon fibres were added to mixtures only before dry pressing (samples 629 from Table 1.) Samples were passed to FTIR examinations. Infrared absorption spectra were taken by BOMEM MB-102 FTIR spectrophotometer equipped with a deuterio-triglicine-sulfate detector, at a resolution of 4 cm⁻¹, in the range of 400-4000 cm⁻¹; 2 mg/g KBr pellets were used. The materials were sintered at 1700 °C in high purity nitrogen by a two-step sinter-HIP method using BN embedding powder. First, some of the samples were sintered without applying pressure. In second step, the samples from first sintering step (serving as reference samples) were re-introduced in the HIP, together with the rest of examined samples. Then, pressure of 20 bar was applied for one hour.

The dimensions of the as-sintered specimens were approximately 3.5 × 5 × 50 mm.

After sintering the weight-gain values were determined. The density of the as-sintered materials was measured by the Archimedes method. To identify the crystalline phases X-ray diffraction of Cu
Kα radiation was applied. Morphology of the solid products was studied by scanning electron microscopy, with a JEOL-25 microscope. The elastic modulus and the four-point bend strength were determined by a bending test with spans of 40 mm and 20 mm. Three-point strength was measured on broken pieces with a span of 20 mm.

The compositional analysis of sintered samples was performed by prompt-gamma activation analysis (PGAA) which is based on the detection of prompt gamma rays originating from (n, γ) reaction, and gives average elemental composition of the total volume of the sample [8].

The composition of starting powder mixtures.

| Batch | Composition, wt% | Added carbon |
|-------|------------------|--------------|
|       | Si₃N₄ | Al₂O₃ | Y₂O₃ | carbon black | graphite | carbon fibre |
| 642   | 90    | 4     | 6    | –          | –        | –          |
| 644   | 90    | 4     | 6    | 3          | –        | –          |
| 645   | 90    | 4     | 6    | –          | 3        | –          |
| 629   | 90    | 4     | 6    | –          | –        | 1          |

3. Results and discussion

Infrared spectroscopy measurements are presented in Fig. 1. Samples extracted from 644 batch during long duration milling experiment can be followed in Fig. 1c, Fig. 1d and Fig. 1e. Infrared spectra of yttria and alpha silicon nitride starting powders are presented in Fig. 1a and Fig. 1b. In yttria powder spectra we can distinguish two characteristic peaks at 606 and 631 cm⁻¹. The mixture of powders mechanically activated for 1 hour (Fig. 1c) is characterized mainly by alpha silicon nitride vibration modes (as in Fig. 1b). O-H stretching modes at 3432 cm⁻¹ and O-H bending at 1634 cm⁻¹ can be found in infrared spectra.

These vibrations are assigned with N-H vibrations in case of inert atmosphere working conditions. At the end of mechanical activation (Fig. 1d) as a result we have a mixture with dominant alpha silicon nitride vibration modes. The vibration modes considered to be Si-N at 600 cm⁻¹ and yttria at 640 cm⁻¹ appeared. At the end of activation O-H bonds disappeared from spectra. After oxidation at 800°C for 2 hours (Fig. 1e) peaks characteristic to alpha silicon nitride and yttria can be observed on infrared spectra. Alumina which has a broad band at 798 cm⁻¹ (not presented in figure) and presents 4 wt% of the mixture can not be seen and has no substantial effect on the spectra of mixtures.

We found the same characteristic vibrations in the case of preparations with graphite (batch 645). After performing the mechanical activation of powder mixtures, rectangular samples were obtained by dry pressing at 220 MPa. The as-obtained samples were oxidized at different temperatures. Oxidizing at different temperatures resulted in samples with different carbon content as presented in Fig. 2. An interesting remark can be added to observations presented in Fig. 2. The behavior of nanocrystalline carbon black and graphite added to silicon nitride matrix was found to be sharply different regarding the oxidation process.

From 450 °C up to 600 °C the carbon black content has a decreasing tendency, at 600 °C was 0.4 wt%. At this stage, the graphite content is around 10 wt%. From this point the graphite content has also a decreasing tendency with increasing tempera-
ture. Continuing the oxidation process above 800 °C, according to weight losses, we obtained a carbon free structure.

Morphological study was performed after two-step sintering by scanning electron microscopy. In Fig. 3a, the microstructure of a reference sample (642) consisting of equiaxial grains can be seen. The structure of the sample characterized by high content of nanocrystalline carbon black is sharply different, the grains are connected through elongated necks (Fig. 3b). Huge sticking grains can be also observed, which developed during a milling process. On the microstructure of sample with no carbon content, derived from oxidation at 800 °C of carbon black containing sample, more developed grains can be observed (Fig. 3c).

Similarly developed grains and sticking tendency can be followed in the case of a sample with high graphite content (Fig. 3d).

A morphological study was made on fractured surfaces as shown in Fig. 4. In Fig. 4a the homogenous microstructure of a reference sample (642) can be seen. From these micrographs can be foreseen that the fracture, crack propagation in this sample involves a combination of intra- and intergranular path which needs higher energy demand. As regards the samples from batch 644 however, the sticking grains of 10–30 μm in size (Fig. 4b, and 4c) act as inclusions in structure and induce intergranular fractures, which requires less energy input.

In Fig. 5a and Fig. 5b carbon fibers pulled out from the matrix and embedded in the matrix can be seen. A deterioration of carbon fibers, cavities developing on the surface of fibers should be noticed.

Relation between apparent density of samples containing carbon (644 and 645) and reference (642) carbon free sample and modulus of elasticity after sintering are presented in Fig. 6.

A higher apparent density and higher modulus than samples with carbon content characterize the 642 reference samples. Samples 644 (nanocrystalline carbon black) have higher modulus values than samples with graphite (645). Linearly fitted straight line 176 were added to Fig. 6 from Ref. 9.
These lines were produced to study partial and final sintering processes. From Fig. 6 it seems that carbon content variation has the same role as partial sintering. The same curves (regression lines) as in the case of line 176, but different gradients can be observed at samples 644 and 645. The linear regression line 176 has the intersection with x axis is at 1.88 value. As conclusion can be drawn that samples 644 at low densities have higher modulus values than the reference samples. This means that at the initial stage of sintering a certain amount of carbon addition has a beneficial role to the modulus of elasticity. This tendency is maintained till 2.338 value obtained from intersection of regression lines 176 and 644. Above this value, although modulus of batches 644 and 645 have an increasing tendency, compared with reference 176 carbon content has a detrimental role to modulus.

In Fig. 7a the x-ray diffractogram of a reference sample can be seen after sintering at 1700 °C, nitrogen atmosphere and without applying pressure. The structure consists of $\beta$-Si$_3$N$_4$ and $\alpha$-Si$_3$N$_4$. The reflection of $\beta$-Si$_3$N$_4$ can be recognized at $d = 0.3041$ nm, however, an unidentified peak has appeared. At this stage of study identification of this peak is still uncertain, however this reflection is close to $\delta$-Y$_2$Si$_2$O$_7$ reflections. In Fig. 7c the reflections of sample of batch 644 can be seen after first sintering step. The structure consists of $\beta$-Si$_3$N$_4$, $\alpha$-Si$_3$N$_4$ and a minor contribution of SiC (JCPDS 31-1231) can be observed. In addition to this reflections at $d = 0.3613$ nm and $d = 0.3041$ nm can be observed, which may be attributed to $\delta$-Y$_2$Si$_2$O$_7$ reflections. After the second sintering step the structure is converting to $\beta$-Si$_3$N$_4$ phase, but retains some of the possible $\delta$-Y$_2$Si$_2$O$_7$ reflections ($d = 0.3613$ nm and $d = 0.3041$ nm) and some of the $\alpha$-Si$_3$N$_4$ reflections ($d = 0.2259$ nm, 0.2619 nm and 0.280 nm).

Fig. 5 Scanning electron micrographs of fracture surface of the carbon fiber containing samples (629) after sintering process. a - fibers pulled out from matrix. b - fibers embedded in matrix, presenting deterioration (holes) on surface.

Fig. 6 Relation between apparent density and modulus of elasticity after sintering. 644 and 645 samples containing carbon, 642 and 176 carbon free samples.

Fig. 7 X-ray diffractograms of one and two step sintered samples; a - reference sample (642) sintered without pressure; b - reference sample (642) sintered under pressure; c - nanocrystalline carbon added (644) sample first step sintered; d - carbon added sample (644) after two-step sintering. Black points mark the new phase(s).

After the second sintering step (Fig. 7b) mostly the reflection of $\beta$-Si$_3$N$_4$ can be recognized. At $d = 0.3041$ nm however, an unidentified peak has appeared. At this stage of study identification of this peak is still uncertain, however this reflection is close to $\delta$-Y$_2$Si$_2$O$_7$ reflections. In Fig. 7c the reflections of sample from batch 644 can be seen after first sintering step. The structure consists of $\beta$-Si$_3$N$_4$, $\alpha$-Si$_3$N$_4$ and a minor contribution of SiC (JCPDS 31-1231) can be observed. In addition to this reflections at $d = 0.3613$ nm and $d = 0.3041$ nm can be observed, which may be attributed to $\delta$-Y$_2$Si$_2$O$_7$ reflections. After the second sintering step the structure is converting to $\beta$-Si$_3$N$_4$ phase, but retains some of the possible $\delta$-Y$_2$Si$_2$O$_7$ reflections ($d = 0.3613$ nm and $d = 0.3041$ nm) and some of the $\alpha$-Si$_3$N$_4$ reflections ($d = 0.2259$ nm, 0.2619 nm and 0.280 nm).

Fig. 8 Relation between carbon content and four point bending strength (BS4) and three point bending strength (BS3). Carbon contents for samples 644 and 645 resulted from oxidation process as in Fig. 3. Samples 629 are characterized by 1 wt% carbon content as in Table 1.
A comprehensive view about the carbon content effect to strength can be seen in Fig. 8. Samples with added carbon (629, 644, 645) present lower values for strength as compared with 642 reference samples.

The compositions of sintered samples were determined by prompt-gamma activation analyses. Comparing the Si/N mass fraction of the reference sample and the composite sample (sample from batch 644, with 11.9 wt% carbon content after oxidation at 450 °C) the effect of carbon content on the complex sintering process is observed, namely in the presence of carbon, the Si/N mass fraction has decreased. Because of high relative uncertainty the amount of oxygen was neglected.

4. Conclusion
Preparation of C/Si3N4 nanocomposites was performed. The milling product of added carbon black and graphite could be characterized by the same structural characteristics, nearly by the same vibration modes. Carbon addition has the same role as partial sintering. In the initial stage of sintering the carbon addition has a beneficial role to the modulus of elasticity. Above a certain value however, carbon content has a detrimental role to the modulus. The morphological studies showed the sticking tendency of powders. Carbon added samples are characterized by lower bending strengths than reference monolithic samples. During a pressureless sintering step the structure retains the α-Si3N4 phase, after the second sintering step the alpha silicon nitride to beta silicon nitride phase transformation was completed. The Si/N fraction decreased during sintering process because of the presence of carbon.

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The composition of sintered samples

| Z    | El | M   | atomic % wt% rel. unc % | Z    | El | M   | atomic % wt% rel. unc % |
|------|----|-----|------------------------|------|----|-----|------------------------|
| 1    | H  | 1.01| 0 0.019 0.9            | 1    | H  | 1.01| 0.438 0.023 7.9        |
| 5    | B  | 10.8| 0.036 0.13 0.9        | 5    | B  | 10.8| 0.438 0.023 7.9        |
| 7    | N  | 14  | 56.05 37.62 1.9       | 6    | C  | 12  | 23.33 14.677 12.8      |
| 13   | Al | 27  | 1.588 2.053 2.9       | 7    | N  | 14  | 38.21 28.028 2.3       |
| 14   | Si | 28.1| 41.16 55.38 2.3       | 13   | Al | 27  | 12.03 16.995 2.3       |
| 39   | Y  | 889 | 1.154 4.915 1.9       | 14   | Si | 28.1| 25.15 36.998 2.6       |
|      |    |     | 889 0.678 3.156       |      |    | 889 | 3.156 2.4             |

Si/N = 1.47 ± 0.04 Si/N = 1.32 ± 0.04