The Uses of 2-Ethoxy-(4H)-3,1-benzoxazin-4-one in the Synthesis of Some Quinazolinone Derivatives of Antimicrobial Activity

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Abstract: The behavior of 2-ethoxy-(4H)-3,1-benzoxazin-4-one (1) towards nitrogen nucleophiles, e.g. ethanolamine, aromatic amines (namely: p-toluidine, p-anisidine, p-hydroxyaniline, o-hydroxyaniline, o-bromoaniline, o-phenylenediamine, p-phenylene-diamine, o-tolidinediamine) p-aminobenzoic acid, glucosamine hydrochloride, 2-amino- nicotinic acid, 1-naphthalenesulfonic acid hydrazide, n-decanoic acid hydrazide, benzoic acid hydrazide, semicarbazide, aminoacids (e.g. D,L-alanine, L-asparagine, L-arginine) and derivatives of 2-aminothiazole has been investigated. The behavior of the benzoxazinone towards a selected sulfur nucleophile, L-cysteine, has also been discussed. Formation of an amidine salt as a reaction intermediate has been assumed. The effect of solvent in some reactions has been elucidated. The structures of all the novel quinazoline and quinazolinone derivatives, obtained by heterocyclic ring opening and ring closure were inferred by the IR, MS as well as 1H-NMR spectral analysis. Moreover, the antimicrobial potential of some of the new synthesized derivatives has been evaluated.

Keywords: benzoxazinone; quinazolinone; aminothiadiazole; nucleoside; antimicrobial

1. Introduction

3,1-Benzoxazin-4-ones can be considered as semiacid anhydrides which undergo many of the reactions of true acid anhydrides, but at a slower rate. This special reactivity allows this class of compounds to be useful as antimicrobial [1], anti-platelet aggregation [2], human leukocyte elastase inhibitors [3], receptor agonist active [4], receptor antagonist active [5-9], pesticides [10], tissue culture protective and in vivo
model of neurodegeneration [11] and improve the umbilical vein endothelial cells [12]. In this paper, we report both the uses of 2-ethoxy-(4H)-3,1-benzoxazin-4-one (1, Scheme 1) in the synthesis of some quinazolinone derivatives and the screening of antimicrobial activity of some of the newly synthesized derivatives against Gram-negative and Gram-positive bacteria as well as fungi by means of the disc diffusion method.

2. Results and Discussion

Herein we report the behavior of benzoxazinone derivative 1 towards some nitrogen and sulfur nucleophiles with the aim of obtaining more precise information about the course of the reaction (the ethoxy group has a small size and negative inductive effect). Thus the reaction of derivative 1 with ethanolamine in boiling ethanol yielded compound 2 (Scheme 1) which on heating above its melting point (120-121 °C) yielded the desired product 3.

Scheme 1. Synthetic pathway for compounds 2 and 3.

The elemental analyses and spectroscopic data for 2 and 3 are consistent with the assigned structures. Isolation of product 2 ruled out the abnormal nucleophilic addition to C-2 to form the amidine salt which subsequently dehydrates to give the desired product 3 [12] (Scheme 2):

Scheme 2. Formation of the amidine salt of compound 3.

The mass spectrum for product 3 showed a molecular ion peak at \( m/z \) 234, 236 for the parent compound followed by ions at \( m/z \) 190,192 and \( m/z \) 174,176 attributable for 2-ethoxyquinazolin-4-one and 2-ethoxyquinazoline, respectively.

Reacting compound 1 with aromatic amines namely, \( p \)-toluidine, \( p \)-anisidine, \( p \)-hydroxyaniline, \( o \)-hydroxyaniline and \( o \)-bromoaniline in boiling ethanol afforded the corresponding 4a-e in good yields. These, in turn, can be cyclized to the corresponding quinazolinones 5a-e under thermal conditions (240-260 °C) or with acetic anhydride. The formation of compound 4 possibly takes place via heteroring opening via nucleophilic addition at the more reactive C-4 in the oxazinoine moiety (no ring closure takes place under this condition due to the small size of ethoxy group that does not enhance cyclization), but under thermal conditions ring closure was obtained (Scheme 3).
Interpretation of the mass spectra for products 5a-e always showed molecular ion peaks of the parent compounds, followed by ion peaks for further fragments, including those for 2-ethoxy-4-(3H)-oxoquinazoline at \( m/z \) 191,193 which could be attributed for the loss of sugar residue, and ending with those for the 2-ethoxyquinazoline at \( m/z \) 175, 177 (Scheme 4).

Reacting compound 1 with \( p \)-aminobenzoic acid (in boiling butanol) and \( p \)-phenylenediamine (in boiling ethanol) afforded quinazolones 6a and 6b respectively. Also, reacting compound 1 with 2-aminonicotinic acid and 3,3'-dimethyl-4,4'-biphenyldiamine (\( o \)-tolidinediamine) each in boiling ethanol afforded quinazolinones 6c and 6d, respectively. But when compound 1 was reacted with \( o \)-phenylenediamine in boiling ethanol it yielded the tetracyclic product 7. The structures of products 6–7 were based on the microanalytical and spectral data (Scheme 5).

Interpretation for the mass spectra for compounds 6a-d and 7 is shown in Scheme 6.
Scheme 6. Mass spectra interpretation for compounds 6a-d and 7.

It is well known that cyclic and acyclic nucleosides often enhance the biological activity of heterocyclic derivatives [11]. Thus, product 1 when glycosidated by glucosamine hydrochloride in the presence of pyridine, afforded the quinazolinone derivative 8 (Scheme 7). The structure of compound 8 was established according to its microanalytical and spectroscopic data.
Scheme 7. Synthetic pathway for compound 8.

The mass spectrum interpretation for compound 8 is interpreted in Scheme 8.

Scheme 8. Mass spectrum interpretation for compound 8.

Sulfonamides are well known for their interesting antibacterial and antifungal activities [13]. Therefore, product 1 when reacted with o-naphthalenesulfonyl hydrazide in boiling ethanol gave product 9 (Scheme 9).

Scheme 9. Synthetic pathway for compound 9.

The mass spectrum interpretation for compound 8 is shown in Scheme 10.

Scheme 10. Mass spectrum interpretation for compound 9.

The reaction of n-decanohydrazide with product 1 in ethanol gave product 10 (Scheme 11).
Scheme 11. Synthetic pathway for product 10.

![Scheme 11](image)

The reaction possibly took place via hydrazide hydrogen that bonded to the nitrogen atom of the heterocycle then underwent an "abnormal" nucleophilic attack to C-2 to form the amidine salt which subsequently dehydrated to give the product (Scheme 12). The structure of product 10 was based on the microanalytical and spectral data.

Scheme 12. Formation of amidine salt of compound 10.

![Scheme 12](image)

An interpretation of the mass spectrum data for compound 10 is shown in Scheme 13.

Scheme 13. Mass spectrum interpretation for compound 10.

![Scheme 13](image)

The outcome of the reaction of compound 1 with benzohydrazide depended on the solvent used. For instance, when the reaction was carried out in n-butanol derivative 11 was obtained, whereas when the reaction occurred in benzene derivative 12 was obtained (Scheme 14).

Scheme 14. Synthetic pathway for compounds 11 and 12.

![Scheme 14](image)

This revealed that in n-butanol the reaction took place via the heteroring opening at C-2 followed by cyclisation to give product 11, where no quinazolone derivative was obtained due to the amidine salt present in the more thermodynamically stable Z-form (Scheme 15), whereas in benzene the product 12 (as the kinetically controlled product) was obtained. However, carrying out this reaction in
ethanol, instead, increases the difficulty for ring closure. This might be because of the lower boiling point of ethanol (78 °C), as a protic solvent, compared with that of water (100 °C), whereas benzene (b.p. 80 °C) is aprotic. The structures of products 11 and 12 were based on microanalytical and spectral data. Interpretation of the mass spectral data is shown in Scheme 16.

**Scheme 15.** Stability of amidine salts forming compounds 11 and 12.

**Scheme 16.** Mass spectra interpretation for compounds 11 and 12.

Heating of compound 1 with semicarbazide in acetic acid in the presence of fused sodium acetate gave compound 13 (Scheme 17). The reaction took place via heteroring opening at C-4 then dehydration. The structure of product 13 was based on microanalytical and spectral data. The mass spectrum interpretation is shown in Scheme 18.

**Scheme 17.** Synthetic pathway for compound 13.
Scheme 18. Mass spectrum interpretation for compound 13.

Benzoxazinone 1 when reacted with amino acids, namely D,L-alanine, L-asparagine, and L-arginine under fusion conditions at 190 °C or by refluxing in pyridine in the presence of few drops of water produced derivatives 14a-c, respectively [14-16] (Scheme 19). The reaction took place via heteroring ring opening at C-4 followed by water elimination. In contrast, the reaction of L-cysteine with compound 1 in boiling pyridine yielded compound 15, where the reaction took place via the heteroring opening at C-4 by the sulfur nucleophile (rather than nitrogen nucleophile) to give derivative 15 as an S-aroylcysteine. The S-substituted cysteines are either products or used as intermediates in many syntheses [17]. The structures of products 14a-c and 15 were based on microanalytical and spectral data.

Scheme 19. Synthetic pathway for compounds 14 and 15.

(i) (a) DL - alanine; (b) L - asparagine; (c) L - arginine; (ii) L - cysteine.

The mass spectra interpretation for compounds 14a-c and 15 is shown in Scheme 20.

Scheme 20. Mass spectra interpretation for compounds 14 and 15.
Reacting benzoxazinone 1 with heteroaromatic amines, namely 2-phenyl-5-aminothiadiazole, 2-cinnamyl-5-aminothiadiazole, or 5-phthalimidomethyl-2-aminothiadiazole in boiling acetic acid in the presence of fused sodium acetate afforded 2-ethoxy-3-substituted-quinazolones 16a-c (Scheme 21). The structures of products 16a-c were based on microanalytical and spectral data. Interpretation for the mass spectra of derivatives 16a-c is shown in Scheme 22.

**Scheme 21.** Synthetic pathway for compound 16.

**Scheme 21.** Mass spectra interpretation for compounds 16a-c.
3. Antimicrobial Evaluation

Compounds 3, 6b, 6d, 8, 9, 10, 12, 14, 16a, 16b and 16c were tested for antimicrobial activity against *Escherichia coli* (Gram negative bacterium), *Staphylococcus aureus* (Gram positive bacterium), *Aspergillus flavus* and *Candida albicans* (fungi) using the disc diffusion method. The antimicrobial evaluation was done in the Microanalytical Center at Cairo University.

3.1. General disc diffusion (agar-based) method

Standard discs of tetracycline (antibacterial agent) and amphotericin B (antifungal agent) served as positive controls and references for antimicrobial activities respectively, but filter discs impregnated with 10 µL of solvent (chloroform, ethanol, DMF) were used as a negative control. The agar used is Meuller–Hinton agar that is rigorously tested for composition and pH. The depth of the agar in the plate is a factor to be considered in this method. Blank paper discs (Schleicher and Schuell, Spain) with a diameter of 8.0 mm were impregnated with 10 µL of the tested concentration of the stock solutions. When a filter paper disc impregnated with a tested chemical is placed on agar, the chemical will diffuse from the disc into the agar. This diffusion will place the chemical in the agar only around the disc. The solubility of the chemical and its molecular size will determine the size of the area of chemical infiltration around the disc. If an organism is placed on the agar it will not grow in the area susceptible to the chemical around the disc. This area of no growth around the disc is the "zone of inhibition" or "clear zone". For disc diffusion, the zone diameters were measured with slipping calipers of the National Committee for Clinical Laboratory Standards (NCCLS) [18]. Agar-based method is a good alternative method being simpler and faster than broth–based methods [19, 20].

4. Antibacterial Activity

Results of antibacterial activity tested against *Escherichia coli* (G-) and *Staphylococcus aureus* (G+) showed that all of the selected compounds are antibacterially active and comparatively efficient.

5. Antifungal Activity

Results of antifungal activity tested against two strains of fungi namely, *Aspergillus flavus* and *Candida albicans* showed that compounds 9, 16b and 16c were active against both fungi, compound 16a was active only with *A. flavus* and compounds 6b, 8, 10 and 12 were only active on *C. albicans*, whereas the rest of compounds were inactive.

| Compound                     | Inhibition zone diameter (mm / mg sample) |
|------------------------------|------------------------------------------|
|                              | E. coli | S. aureus | A. flavus | C. albicans | Solvent | control |
| Tetracycline                 | 33      | 31        | 00        | 00          | ----    |
| Amphotericin B               | 00      | 00        | 17        | 21          | ----    |
| 3                            | 12      | 12        | 00        | 12          | Chloroform |
| 6b                           | 22      | 17        | 00        | 15          | Ethanol |
| 6d                           | 15      | 14        | 00        | 00          | Ethanol |
Table 1. Cont.

| Compound No. | Inhibition Zone Diameter (mm) |
|--------------|------------------------------|
| 3            | 13                           |
| 6b           | 16                           |
| 6d           | 16                           |
| 8            | 15                           |
| 9            | 12                           |
| 10           | 00                           |
| 12           | 00                           |
| 14b          | 00                           |
| 16a          | 17                           |
| 16b          | 15                           |
| 16c          | 10                           |

In conclusion all the compounds 3, 6b, 6d, 8, 9, 10, 12, 14b and 16a-c were antibacterially active and comparatively efficient. In addition, compounds 9, 16b and 16c were active against both fungi, 16a was active only with A. flavus, 6b, 8, 10 and 12 were active only with C. albicans, and the rest were inefficient. The following graph (Figure 1) represents the antimicrobial activity of these products compared to those of tetracycline and Amphotericin B.

**Figure 1.** Graphical representation of the antimicrobial activity of tested compounds compared to tetracycline and amphotericin B.

6. Experimental

6.1. General

All melting points recorded are uncorrected. The IR spectra were recorded on a Pye Unicam SP1200 spectrophotometer using the KBr wafer technique. The $^1$H-NMR spectra were determined on a Varian FT-200 or Bruker AC-200 MHz instrument using TMS as an internal standard. Chemical shifts ($\delta$) are expressed in ppm. The mass spectra were determined using MP model NS-5988 and Shimadzu single focusing mass spectrometer (70 eV). All solvents used were of HPLC/AnalaR grade. All reagents were used as received from Alfa Aesar. Compound 1 was prepared according to methods available in the
literature [21], and in the order to avoid moisture it was immediately used after preparation, prior to each synthesis.

6.2. General Procedure for the Synthesis of Compounds 2 and 3

A mixture of 2-ethoxy(4H)-3,1-benzoxazin-4-one 1 (0.01 mol) and ethanolamine (0.01 mol) in boiling ethanol (30 mL) was refluxed for 3 h. Concentration of the solvent left a white precipitate of compound 2 which was crystallized from ethanol affording beige white crystals. Heating product 2 above its melting point yielded the corresponding product 3.

2-Ethoxycarbonylamino(β-hydroxyethyl)benzamide (2): Yellowish white crystals from ethanol; m.p. 120-121 °C; yield 80%; Anal. for C_{12}H_{16}N_{2}O_{4} (m.w. 252); Found: C, 57.21; H, 6.29; N, 11.13; Calcd: C, 57.14; H, 6.35; N, 11.11; IR ν (cm⁻¹) 1636 (C=O), 1737 (C=O), 3069 (CH), 3130 (NH), 3342 (OH); MS: m/z [252]⁺.

2-Ethoxycarbonylamino–N-(4-methylphenyl)benzamide (4a): Brown crystals from ethanol; m.p. 116–117 °C; yield 85%; Anal. for C_{17}H_{18}N_{2}O_{3} (m.w. 298); Found: C, 68.68; H, 6.12; N, 9.40; Calcd: C, 68.46; H, 6.04; N, 9.40; IR ν (cm⁻¹) 1636, 1726 (2xC=O), 3361 (NH). MS: m/z (int. %) [M+H]⁺ 298 (236 (12.8), 100 (100), 192 (22.3); 1H-NMR (DMSO-d₆) δ 1.28 (t, 3H; -OCH₂CH₃, J = 7.4 Hz), 2.21 (s, 3H; ArCH₃), 4.16 (q, 2H; -OCH₂CH₃, J = 7.4 Hz), 7.08–7.18 (m, 4H; Ph-H), 6.99–7.67 (m, 4H, quinazolinone), 9.92 (s, 1H, amide NH), 10.76 (s, 1H, amino, D₂O exchangeable).

2-Ethoxycarbonylamino–N-(4-methoxyphenyl)benzamide (4b): Brown crystals from ethanol; m.p. 133–134 °C; yield 80%; Anal. for C_{17}H_{18}N_{2}O_{4} (m.w. 314); Found: C, 64.88; H, 5.64; N, 8.89; Calcd: C, 64.97; H, 5.73; N, 8.91; IR ν (cm⁻¹) 1638, 1725 (2xC=O), 3361 (NH). MS: m/z (int. %) [M+H]⁺ 314 (68.2%).

2-Ethoxycarbonylamino–N-(4-hydroxyphenyl)benzamide (4c): Brown crystals from ethanol; m.p. 114–115 °C; yield 75%; Anal. for C_{16}H_{16}N_{2}O_{4} (m.w. 300); Found: C, 64.06; H, 5.28; N, 9.21; Calcd: C, 64.00; H, 5.33; N, 9.33; IR ν (cm⁻¹) 1640, 1734 (2xC=O), 3321 (NH), 2992 (OH).
2-Ethoxycarbonylamino-N-(2-hydroxyphenyl)benzamide (4d): Brown crystals from ethanol; m.p. 111-112 °C; yield 75%; Anal. for C_{16}H_{16}N_{2}O_{4} (m.w. 300); Found: C, 64.02; H, 5.30; N, 9.23; Calcd: C, 64.00; H, 5.33; IR ν (cm^{-1}) 1638, 1713 (2xC=O), 3268 (NH), 3402 (OH). MS: m/z (int. %) [M+H]^+ 300 (44%).

2-Ethoxycarbonylamino-N-(2-bromophenyl)benzamide (4e): Brown crystals from ethanol; m.p. 105-106 °C; yield 85%; Anal. for C_{16}H_{15}BrN_{2}O_{3} (m.w. 363); Found: C, 52.77; H, 4.16; N, 7.72; Br, 22.16; Calcd: C, 52.89; H, 4.13; N, 7.71; Br, 22.04; IR ν (cm^{-1}) 1637, 1740 (2xC=O), 3402 (NH); MS: m/z (int.%) [M+H]^+ 363 (47.4%).

6.4. General procedure for the synthesis of compounds 5a-e

Method A: Compounds 4a-e (0.01 mol) were heated in a round bottom flask (25 mL) in an oil bath at 160 °C for 30 minutes. After cooling the products were crystallized from the proper solvent to give the corresponding quinazolinones 5a-e, respectively.

Method B: A solution of compound 4a-e (0.01 mol) in acetic anhydride (15 mL) was heated under reflux for 2 hr. The solid that separated out after dilution with water was filtered off, dried and crystallized from the proper solvent to give compounds 5a-e, respectively.

2-Ethoxy-3-(4-methylphenyl)quinazolin-4-one (5a): Dark brown crystals from light petroleum (100–120 °C); m.p. 91 – 92 °C; yield 90%; Anal. for C_{17}H_{16}N_{2}O_{2} (m.w. 280); Found: C, 72.34; H, 5.62; N, 9.89; Calcd: C, 72.34; H, 5.67; N, 9.93; IR ν (cm^{-1}) 1670 (C=O); MS: m/z (int. %) [M+H]^+ 280 (58.0), 282 (13.8), 191 (100), 193 (18.2), 175 (8.3), 177 (0.7), 157 (6.3), 159 (0.6), 130 (67.0), 132 (11.9); ^1H-NMR (DMSO-d$_6$) δ 1.24 (t, 3H; -OCH$_2$CH$_3$, J = 7.4 Hz), 2.21 (s, 3H; ArCH$_3$), 4.18 (q, 2H; -OCH$_2$CH$_3$, J = 7.4 Hz), 7.40–7.60 (m, 4H; Ph-H), 7.08–7.18 (m, 4H, quinazolinone).

2-Ethoxy-3-(4-methoxyphenyl)quinazolin-4-one (5b): Dark brown crystals from toluene; m.p. 114 – 115 °C; yield 85%; Anal. for C_{17}H_{16}N_{2}O_{3} (m.w. 298); Found: C, 68.32; H, 5.32; N, 9.41; Calcd: C, 68.46; H, 5.37; N, 9.40; IR ν (cm^{-1}) 1668 (C=O); MS: m/z (int. %) [M+H]^+ 298 (52.3), 300 (31.1), 191 (100), 193 (13.5), 175 (27.2), 177 (0.7), 157 (3.3), 159 (0.4), 130 (54.6), 132 (8.3); ^1H-NMR (DMSO-d$_6$) δ 1.22 (t, 3H; -OCH$_2$CH$_3$, J = 7.4 Hz), 3.76 (s, 3H; OCH$_3$), 4.35 (q, 2H; -OCH$_2$CH$_3$, J = 7.4 Hz), 6.66–7.68 (m, 4H; Ph-H), 7.47–8.19 (m, 4H, quinazolinone).

2-Ethoxy-3-(4-hydroxyphenyl)quinazolin-4-one (5c): Dark brown crystals from benzene; m.p. 105–106 °C; yield 80%; Anal. for C_{16}H_{14}N_{2}O_{3} (m.w. 284); Found: C, 67.42; H, 4.70; N, 9.81; Calcd: C, 67.60; H, 4.93; N, 9.86; IR ν (cm^{-1}) 1671 (C=O), 2992 (OH); MS: m/z (int. %) [M+H]^+ 284 (43.0), 286 (16.3), 191 (100), 193 (52.3), 175 (20.8), 177 (0.1), 157 (4.3), 159 (0.5), 130 (57.4), 132 (7.9); ^1H-NMR (DMSO-d$_6$) δ 1.22 (t, 3H; -OCH$_2$CH$_3$, J = 7.4Hz), 4.35 (q, 2H; -OCH$_2$CH$_3$, J = 7.4 Hz), 5.35 (s, H; OH), 6.68–7.69 (m, 4H; Ph-H), 7.43–8.19 (m, 4H, quinazolinone).

2-Ethoxy-3-(2-hydroxyphenyl)quinazolin-4-one (5d): Dark brown crystals from light petroleum (100–120 °C); m.p. 89–90 °C; yield 80%; Anal. for C_{16}H_{14}N_{2}O_{3} (m.w. 284); Found: C, 67.31; H, 4.73; N, 9.89; Calcd: C, 67.60; H, 4.93; N, 9.86; IR ν (cm^{-1}) 1669 (C=O), 2988 (OH); MS: m/z (int. %) [M+H]^+
284 (48.0), 286 (6.3), 191 (100), 193 (43.3), 175 (18.2), 177 (0.1), 157 (4.7), 159 (1.2), 130 (67.1), 132 (6.3); 1H-NMR (DMSO-d6) δ 1.22 (t, 3H; -OCH₂CH₃, J = 7.4 Hz), 4.36 (q, 2H; -OCH₂CH₃, J = 7.4 Hz), 5.55 (s, H; OH), 6.76–7.67 (m, 4H; Ph-H), 7.30–8.19 (m, 4H; quinazolinone).

2-Ethoxy-3-(2-bromophenyl)quinazolin-4-one (5e): Dark brown crystals from light petroleum (80 – 100 °C); m.p. 93 – 94 °C; yield 90%; Anal. for C₁₆H₁₃BrN₂O₂ (m.w. 345); Found: C, 67.31; H, 4.73; N, 9.89; Br, 23.25; Calcd: C, 67.60; H, 4.9; N, 9.86; Br, 23.19; IR ν (cm⁻¹) 1670 (C=O); MS: m/z (int. %) [M+H]+ 345 (66.0), 347 (22.4), 191 (100), 193 (28.3), 175 (23.3), 177 (0.2), 157 (5.8), 159 (0.4), 130 (61.4), 132 (5.4); 1H-NMR (DMSO-d6) δ 1.21 (t, 3H; -OCH₂CH₃, J = 7.4 Hz), 4.36 (q, 2H; -OCH₂CH₃, J = 7.4 Hz), 7.26-7.51 (m, 4H; Ph-H), 7.29-8.19 (m, 4H, quinazolinone). Compounds 5a-e were devoid any ester and/or NH group bands.

6.5. General Procedure for the Synthesis of Compounds 6a-d

A mixture of benzoxazinone 1 (0.01 mol) and any of the aromatic amines p-aminobenzoic acid, 2-aminonicotinic acid, p-phenylenediamine, or 3,3′-dimethyl-4,4′-biphenyldiamine (0.01 mol) in boiling butanol (30 mL) was refluxed for 3-6 h (depending on the nucleophile). Concentrating the solution gave a solid which was washed, filtered, dried and then recrystallized from the proper solvent affording the desired quinazolinone derivatives 6a-d, respectively.

4-[2-Ethoxy-4-quinazolon-3-yl]benzoic acid (6a): Light brown crystals from ethanol; m.p. 151 – 152 °C; yield 80%; Anal. for C₁₇H₁₄N₂O₄ (m.w. 310); Found: C, 65.44; H, 4.72; N, 9.00; Calcd: C, 65.80; H, 4.52; N, 9.03; IR ν (cm⁻¹) 1675, 1705 (2xC=O), 3355 (chelated OH); MS: m/z (int. %) [M+H]+ 310 (63.0), 312 (28.2), 191 (100), 193 (27.8), 175 (34.3), 177 (0.6), 130 (59.3), 132 (0.2), 122 (1.8), 124 (0.2), 78 (0.1), 80 (0.1); 1H-NMR (DMSO-d6) δ 1.22 (t, 3H; -OCH₂CH₃, J = 6.9 Hz), 4.36 (q, 2H; -OCH₂CH₃, J = 6.9 Hz), 7.40-8.00 (m, 4H; Ph-H), 7.43-8.19 (m, 4H, quinazolinone), 10.6 (1H, acid proton).

2-Ethoxy-3-(4-aminophenyl)quinazolin-4-one (6b): Light blue crystal from ethanol; m.p. 105 – 106 °C; yield 80%. Anal. for C₁₆H₁₂N₃O₂ (m.w. 278); Found: C, 69.22; H, 4.16; N, 15.08; Calcd: C, 69.06; H, 4.32; N, 15.10; IR ν (cm⁻¹) 1634 (C=N), 1670 (C=O), 3225 (NH). MS: m/z (int. %) [M+H]+ 278 (55.0), 280 (18.2), 191 (100), 193 (31.7), 175 (32.1), 177 (0.8), 157 (4.7), 159 (0.6), 130 (52.4), 132 (0.3), 93 (0.9), 95 (0.1), 78 (0.2), 80 (0.1); 1H-NMR (DMSO-d6) δ 1.22 (t, 3H; -OCH₂CH₃, J = 6.9 Hz), 6.97, 7.87, 8.41 (m, 3H; H-4, H-5, pyridine moiety H-6).

2-[2-Ethoxy-4-oxoquinazolin-384H)-yl]pyridine-3-carboxylic acid (6c): Brown crystals from ethanol; m.p. 298-300 °C; yield 80%; Anal. for C₁₆H₁₃N₂O₄ (m.w. 311); Found: C, 61.58; H, 4.26; N, 13.20; Calcd: C, 61.74; H, 4.18; N, 13.50; IR ν (cm⁻¹) 1634, 1704 (2xC=O), 3255 (chelated OH); MS: m/z (int. %) [M+H]+ 311 (36.0), 313 (19.5), 191 (100), 193 (28.6), 175 (61.8), 177 (7.6), 157 (5.2), 159 (1.1), 130 (58.2), 132 (0.6), 123 (0.7), 125 (0.2), 79 (0.7), 81 (0.1); 1H-NMR (DMSO-d6) δ 1.22 (t, 3H; -OCH₂CH₃, J = 6.9 Hz), 4.37 (q, 2H; -OCH₂CH₃, J = 6.9), 7.44-8.20 (m, 4H; quinazolinone), 5.97, 7.87, 8.41 (m, 3H; H-4, H-5, pyridine moiety H-6).
2-Ethoxy-3-(3,3'-dimethyl-4-amino)biphenylquinazolin-4-one (6d): Dark brown crystals from ethanol; m.p. 121-122 °C; yield 80%; Anal. for C$_{24}$H$_{23}$N$_{3}$O$_{2}$ (m.w. 385); Found: C, 74.53; H, 5.63; N, 10.79; Calcd: C, 74.80; H, 5.97; N, 10.90; IR (KBr) υ (cm$^{-1}$) 1620 (C=N), 1675 (C=O), 3250 (NH). MS: m/z (int. %) [M+H]$^+$ 385 (72.0), 387 (34.2), 228 (13.3), 230 (1.2), 198 (12.7), 200 (0.1), 194 (16.1), 196 (0.4), 191 (100), 193 (24.8), 175 (53.3), 177 (8.7), 157 (7.1), 159 (1.4), 154 (1.2), 156 (0.2), 130 (46.3), 132 (0.1), 78 (0.3), 80 (0.1). $^1$H-NMR (DMSO-d$_6$) δ 1.22 (t, 3H; -OCH$_2$CH$_3$, $J$ = 6.9 Hz), 2.21-2.29 (s, 6H; 2ArCH$_3$), 4.4 (q, 2H; -OCH$_2$CH$_3$, $J$ = 6.9 Hz), 5.08 (2H, s, NH$_2$), 6.90-7.80 (m, 6H; biphenyl-H), 7.33-8.19 (m, 4H, quinazolinone).

2-Ethoxy benzimidazo-[1,2-c]quinazoline (7): A mixture of benzoxazinone 1 (0.01 mol) and o-phenylenediamine (0.01 mol) in boiling butanol (30 mL) was refluxed for 3 h. Concentrating the solution left a solid product which was filtered, washed, dried and crystallized from ethanol affording light blue crystals of product 7. M.p. 191-192 °C; yield 85%. Anal. for C$_{16}$H$_{13}$N$_{3}$O (m.w. 263); Found: C, 73.02; H, 4.94; N, 15.98; Calcd: C, 73.00; H, 4.94; N, 15.97; IR υ (cm$^{-1}$) 1607 (C=N). MS: m/z (int. %) [M+H]$^+$ 263 (66.0), 265 (17.3), 187(44.8), 189 (4.8), 175 (28.4), 177 (11.1), 157 (3.3), 159 (0.1), 130 (78.9), 132 (12.4), 78 (2.4), 80 (0.1). $^1$H-NMR (DMSO-d$_6$) δ 1.22 (t, 3H; -OCH$_2$CH$_3$, $J$ = 7.1Hz), 4.31(q, 2H; -OCH$_2$CH$_3$, $J$ = 7.1 Hz), 7.52-7.92 (m, 4H, benzimidazole), 7.62-8.55 (2 m, 4H; quinazoline).

2-Ethoxy-3-(β-glucopyranosyl-3-yl) quinazolin-4-one (8): A mixture of benzoxazinone 1 (0.01 mol) and glucosamine hydrochloride (0.01 mol) in pyridine (30 mL) was refluxed for 3 h. The mixture was poured in to an ice / water mixture stirred leaving a white precipitate to settle down. The precipitate was washed with water, filtered, dried and then crystallized from ethanol affording white crystals of product 8. M.p. 179–180 °C; yield 80%. Anal. for C$_{16}$H$_{20}$N$_{2}$O$_7$ (m.w. 352); Found: C, 54.50; H, 5.67; N, 7.99; Calcd: C, 54.54; H, 5.68; N, 7.95; IR υ (cm$^{-1}$) 1675 (C=O), 3236 (OH bonded), 3400 (OH non-bonded). MS: m/z (int. %) [M+H]$^+$ 352 (48.0), 354 (21.1), 337 (19.2), 339 (8.2), 307 (2.8), 309 (1.1), 277 (0.8), 279 (0.2), 233 (0.1), 235 (0.1), 191 (88.1), 193 (3.1), 175 (100), 177 (13.1), 157 (4.6), 159 (0.5), 130 (68.9), 132 (9.4); $^1$H-NMR (DMSO-d$_6$) δ 1.1 (t, 3H; -OCH$_2$CH$_3$, $J$ = 6.8 Hz), 3.21-4.20 (m, 6H; H-3', H-4', H-5', H-6', H-7'a, and H-7'b), 3.7 (m, 2'-OH, 4'-OH, 5'-OH), 4.35 (q, 2H; -OCH$_2$CH$_3$, $J$ = 6.8 Hz), 5.0 (d, 1H; H-2', $J$ = 6.5 Hz), 4.9 (s, 7'-OH), 7.41-8.17 (m, 4H, quinazolinone).

2-Ethoxy-3-(1-naphthylsulphonylamino)-4(3H) quinazolone (9): A mixture of benzoxazinone 1 (0.01 mol) and 1-naphthalenesulfonic hydrazide (0.01 mol) in ethanol (30 mL) was heated under reflux for 3 h. The excess solvent was distilled off and the mixture was cooled down leaving a brown solid, which was recrystallized from ethanol affording white brown crystals of product 9. M.p. 111-112 °C; yield 65%. Anal. for C$_{20}$H$_{17}$N$_{3}$O$_4$S (m.w. 395); Found: C, 60.90; H, 4.16; N, 10.81; S, 8.11; Calcd: C, 60.76; H, 4.30; N, 10.63; S, 8.10; IR υ (cm$^{-1}$) 1160 (S=O), 1610 (C=N), 1675 (C=O), 3200 (NH). MS: m/z (int. %) [M+H]$^+$ 395 (63.2), 397 (13.2), 207 (22.7), 209 (2.4), 191 (100), 193 (15.5), 175 (14.5), 177 (0.4), 157 (3.5), 159 (0.3), 130 (58.4), 132(15.6), 128 (13.6); $^1$H–NMR (DMSO-d$_6$) δ 1.19 (t, 3H; -OCH$_2$CH$_3$, $J$ = 6.8 Hz), 4.44 (q, 2H; -OCH$_2$CH$_3$), 7.29 – 8.20 (m, 4H, quinazolinone), 7.72 – 8.05 (m, 7H, naphthalene); 10.6 (s, NH exchangeable).

2-Ethoxy-3-(I-naphthylsulphonylamino)-4(3H) quinazolone (9): A mixture of benzoaxazinone 1 (0.01 mol) and 1-naphthalenesulfonic hydrazide (0.01 mol) in ethanol (30 mL) was heated under reflux for 3 h. The excess solvent was distilled off and the mixture was cooled down leaving a brown solid, which was recrystallized from ethanol affording white brown crystals of product 9. M.p. 111-112 °C; yield 65%. Anal. for C$_{20}$H$_{17}$N$_{3}$O$_4$S (m.w. 395); Found: C, 60.90; H, 4.16; N, 10.81; S, 8.11; Calcd: C, 60.76; H, 4.30; N, 10.63; S, 8.10; IR υ (cm$^{-1}$) 1160 (S=O), 1610 (C=N), 1675 (C=O), 3200 (NH). MS: m/z (int. %) [M+H]$^+$ 395 (63.2), 397 (13.2), 207 (22.7), 209 (2.4), 191 (100), 193 (15.5), 175 (14.5), 177 (0.4), 157 (3.5), 159 (0.3), 130 (58.4), 132(15.6), 128 (13.6); $^1$H–NMR (DMSO-d$_6$) δ 1.19 (t, 3H; -OCH$_2$CH$_3$, $J$ = 6.8 Hz), 4.44 (q, 2H; -OCH$_2$CH$_3$), 7.29 – 8.20 (m, 4H, quinazolinone), 7.72 – 8.05 (m, 7H, naphthalene); 10.6 (s, NH exchangeable).

2-Ethoxy-3-decanoylamino-4(3H) quinazolinone (10): A mixture of benzoaxazinone 1 (0.01 mol) and decanoic acid hydrazide (0.01 mol) in ethanol (30 mL) was refluxed for 3 h. The solvent was
concentrated and the mixture left to cool. Crystallization of the product from ethanol yielded light brown crystals of product 10. M.p. 115–116 °C; yield 75%. Anal. for C_{20}H_{29}N_{3}O_{3} (m.w. 359); Found: C, 66.83; H, 8.12; N, 11.37; Calcd: C, 66.85; H, 8.08; N, 11.70; IR \nu (cm^{-1}) 1670 (C=O), 3250 (NH). MS: m/z (int. %) [M+H]^+ 359 (77.0), 361 (21.3), 191 (100), 193 (22.4), 175 (13.8), 177 (0.6), 171 (2.8), 173 (0.5), 157 (3.8), 159 (0.2), 143 (1.7), 145 (0.3), 130 (27.3), 132 (13.2); ^1H-NMR (DMSO-d$_6$) \delta 0.78 (t, 3H; CH$_3$), 1.15 (t, 3H; -OCH$_2$CH$_3$), 2.3 (t, 2H; CH$_2$CO), 4.35 (q, 2H; -OCH$_2$CH$_3$), 7.46-8.20 (m, 4H; ArH), 10.7 (s, NH, exchangeable).

2-(3-Ethoxy-5-phenyl-4H-1,2,4-triazolo-4-yl)benzoic acid (11): A mixture of benzoxazinone 1 (0.01 mol) and benzoic acid hydrazide (0.01 mol) in butanol (30 mL) was refluxed for 3 h. The solvent was concentrated leaving a white solid, which was crystallized from ethanol affording white crystals of product 11. M.p. 101-102 °C; yield 80%. Anal. for C$_{17}$H$_{15}$N$_3$O$_3$ (m.w. 309); Found: C, 66.14; H, 4.53; N, 13.17; Calcd: C, 66.02; H, 4.85; N, 13.59; IR \nu (cm^{-1}) 1690 (C=O), 2616-3481 (chelated OH). MS: m/z (int. %) [M+H]^+ 309 (78.0), 311 (3.8), 190 (42.5), 192 (13.1), 146 (28.6), 148 (9.1), 122 (2.2), 124 (0.2), 78 (0.5), 80 (0.1); 70 (0.3), 72 (0.1), 69 (0.2), 71 (0.1); ^1H-NMR (DMSO-d$_6$) \delta 1.16 (t, 3H; -OCH$_2$CH$_3$, J = 7.2 Hz), 4.22 (q, 2H; -OCH$_2$CH$_3$), 7.46–8.00 (5H, m, Ph-H) and 7.28–8.20 (m, 4H, quinazolinone), 11.2 (s, broad 1H; OH).

2-Ethoxy-3-benzoylamino-4(3H) quinazolinone (12): Refluxing the mixture of benzoxazinone 1 and benzoic acid hydrazide (0.01 mol each) in benzene (30 mL) for 3 h gave a white solid product that was filtered, washed, dried and crystallized from ethanol giving white needles of 12 of m.p. 109–110 °C; yield 85%. Anal. for C$_{17}$H$_{15}$N$_3$O$_3$ (m.w. 309); Found: C, 66.02; H, 4.93; N, 13.57; Calcd: C, 66.02; H, 4.85; N, 13.59; IR \nu (cm^{-1}) 1635, 1670 (2C=O), 3230 (NH). MS: m/z (int. %) [M+H]^+ 309 (55.0), 311 (11.8), 191 (100), 193 (13.7), 175 (57.2), 177 (14.2), 157 (3.9), 159 (0.5), 130 (42.5), 132 (13.2), 122 (35.6), 124 (0.1); ^1H-NMR (DMSO-d$_6$) \delta 1.15 (t, 3H; -OCH$_2$CH$_3$), 4.44 (q, 2H; -OCH$_2$CH$_3$), 7.18–8.25 (4Xd, 4H, ArH) and 7.28–8.20 (m, 4H, quinazolinone ArH).

2-Ethoxy-5-oxo-1H-1,2,4-triazolo[2,3-c] quinazoline (13): A mixture of benzoxazinone 1 (0.01 mol) and semicarbazide (0.01 mol) in acetic acid/fused sodium acetate (30 mL/2 g) was refluxed for 3 h. Pouring the solution into an ice/water mixture left a white solid, which was filtered, washed, dried and recrystallized from ethanol affording white crystals of derivative 13. M.p. 126–127 °C; yield 75%. Anal. for C$_{11}$H$_{10}$N$_4$O$_2$ (m.w. 230); Found: C, 57.39; H, 4.36; N, 24.35; Calcd: C, 57.39; H, 4.35; N, 24.35; IR \nu (cm^{-1}) 1671 (C=O), 3206 (bonded NH), 3320 (non-bonded NH). MS: m/z (int. %) [M+H]^+ 230 (77.0), 232 (14.3), 190 (17.3), 192 (8.2), 188 (12.4), 157 (8.4), 159 (1.5), 143 (15.4), 145 (2.3), 130 (58.6), 132 (2.0); ^1H-NMR (DMSO-d$_6$) \delta 1.19 (t, 3H; -OCH$_2$CH$_3$), J = 7.2 Hz), 4.43 (q, 2H; -OCH$_2$CH$_3$), 7.18–8.25 (4Xd, 4H, ArH), 10.5 (s, NH, exchangeable).

6.6. General procedure for the synthesis of 2-(ethoxy) quinazolinone-3-yl-alkylacetic acids 14a-c

Method A: A mixture of benzoxazinone 1 (0.01 mol) and selected amino acids namely, D,L-alanine, L-asparagine, and L–arginine (0.01 mol) was fused in an oil bath at 190 °C for 2 h. The mixture was then poured in an ice/water mixture, stirred and left allowing the white precipitate to settle down. The precipitate was filtered, washed, dried and finally crystallized from the proper solvent.
Method B: A mixture of benzoxazinone 1 (0.01 mol) and selected amino acids namely, D,L-alanine, L-asparagine, and L-arginine (0.01 mol) were refluxed in a pyridine (30 mL)/water (5 mL) mixture for 3 h. The mixture was then poured in an ice/water mixture, stirred and left to allow the white solid precipitate to settle down. The solid was filtered, washed, dried and finally crystallized from the proper solvent to yield compounds 14a-c, respectively.

2-{2-(Ethoxy)quinazolin-4-one-3-yl}propanoic acid (14a): Light brown crystals from DMF; m.p. 183–184 °C; yield (70% by Method A and 85% by Method B). Anal. for C_{13}H_{14}N_{2}O_{4} (m.w. 262); Found: C, 59.59; H, 5.30; N, 10.67; Calcd: C, 59.54; H, 5.34; N, 10.69; IR \nu (cm^{-1}) 1675 (C=O cyclic amide), 1700 (C=O amide), 3300 (chelated OH). MS: m/z (int. %) [M+H]^+ 262 (82.0), 264 (32.1), 234 (22.7), 236 (2.8), 191 (100), 193 (22.4), 175 (9.9), 177(2.7), 157 (6.1), 159 (0.2), 130 (49.3), 132 (18.4); \textsuperscript{1}H-NMR (DMSO-d_6) \delta 1.20 (t, 3H; -OCH_2CH_3, J= 7.1 Hz), 4.35(q, 2H; -OCH_2CH_3), 7.45-8.17 (m, 4H , Ar-H), 10.5 (1H, acid-H).

2-{2-(Ethoxy)quinazolin-4-one-3-yl}-3-carbamoyl propanoic acid (14b): Light brown crystals from DMF; m.p. 241-242 °C; yield (75% by Method A and 85% by Method B). Anal. for C_{14}H_{15}N_{3}O_{7} (m.w. 305); Found: C, 55.16; H, 4.88; N, 13.79; Calcd: C, 55.08; H, 4.92; N, 13.77; IR \nu (cm^{-1}) 1670, 1675, 1705 (3 x C=O amides and acid), 3200 (NH), 3350 (chelated OH). MS: m/z (int. %) [M+H]^+ 305 (80.0), 307 (11.5), 287 (12.3), 289 (1.3), 243 (12.2), 245 (2.3), 190 (100), 192 (32.2), 188 (8.4), 190 (0.8), 175 (48.6), 177 (12.3), 179 (0.7), 157 (3.8), 159 (0.1), 130 (55.2), 132 (0.3); \textsuperscript{1}H-NMR (DMSO-d_6) \delta 1.2 (t, 3H; -OCH_2CH_3, J= 7.1 Hz), 3.17 (m, 2H; CH_2), 3.0 (s, NH), 5.7 (q, H; methine H, J= 6.7 Hz), 4.35 (q, 2H; -OCH_2CH_3), 7.45-8.17 (m, 4H, Ar-H), 10.6 (1H, acid-H).

2-Amino-2-{2-(ethoxy)quinazolin-4-one-3-yl}-5-guanidinopentanoic acid (14c): Light brown crystals from DMF; m.p. 241-242 °C; yield (70% by Method A, 80% by Method B). Anal. for C_{16}H_{21}N_{5}O_{4} (m.w. 347); Found: C, 55.12; H, 6.13; N, 20.16; Calcd: C, 55.33; H, 6.05; N, 20.17; IR \nu (cm^{-1}) 1571(NH_2), 1613 (C=N guanidinium), 1670 (C=O cyclic amide), 1700 (C=O acid), 3181 (NH), 3350 (chelated OH). MS: m/z (int. %) [M+H]^+ 348 (80.0), 350 (34.2), 330 (2.4), 332 (0.2), 312 (3.8), 314 (0.2), 285 (2.2), 287 (0.1), 191 (100), 193 (42.2), 175 (38.1), 157 (4.6), 159 (0.5), 157 (6.7), 159 (0.4), 177 (12.9), 130 (48.2), 132 (0.1); \textsuperscript{1}H-NMR (DMSO-d_6) \delta 1.2 (t, 3H; -OCH_2CH_3, J= 7.1 Hz), 1.69 (2H, m, arg. C-4), 1.95 (2H, m, arg. C-3), 3.43 (2H, t, arg. C-5), 5.39 (1H, t, arg. C-2), 4.38 (q, 2H; -OCH_2CH_3), 7.46 – 8.17 (m, 4H, Ar-H), 6.9-7.1 (br. d, NH_2 guanidinium grp.), 7.8 – 8.0 (br. s., guanidinium grp. NH), 10.5 (1H, acid-H).

2-Ethoxycarbonylamino-(carboxyaminomethyl thiomethyl benzoate) (15): Benzoxazinone 1 and cysteine (0.01 mol each) were refluxed in a mixture of pyridine (30 mL) / water (5mL) for 3h. The mixture was then poured in an ice/water mixture, stirred and left yielding a white solid precipitate. The precipitate was filtered, washed, dried and crystallized from DMF affording white crystals of product 15. M.p.121-122 °C; yield 85%. Anal. for C_{16}H_{16}N_{2}O_{5}S (m.w. 312); Found: C, 50.39; H, 5.22; N, 8.88; S, 10.02; Calcd: C, 50.00; H, 5.13; N, 8.97; S, 10.26. IR \nu (cm^{-1}) 1670, 1735, 3223, 3340. MS: m/z (int. %) [M+H]^+ 312 (82.0), 314 (48.0), 248 (28.8), 250 (11.1), 144 (12.6), 146 (2.3), 122 (0.8), 124 (0.1), 78 (0.2), 80 (0.1); \textsuperscript{1}H-NMR (DMSO-d_6) \delta 1.24 (t, 3H; -OCH_2CH_3, J = 7.2 Hz), 3.56 (d, 2H; CH_2, J = 6.5
Hz), 3.63 (t, methine proton, $J = 6.5$ Hz), 4.19 (q, 2H; -OCH$_2$CH$_3$), 6.5 (1H, NH, exchangeable), 7.29-7.53 (m, 4H; ArH), 10.6 (1H, acid-H).

6.7. General procedure for the synthesis of 2-ethoxy-3-substituted quinazolones 16a-c

A mixture of benzoxazinone 1 (0.01 mol) and the aminothiadiazole derivatives 2-phenyl-5-aminothiadiazole, 2-cinnamyl-5-aminothiadiaze, and 2-phthalimidomethyl-5-aminothiazaole (0.01 mol) was refluxed in boiling acetic acid/fused sodium acetate (30 mL/2 g) for 3 h. The solution was poured into an ice/water mixture, stirred and left to settle down affording a white solid. The resulting solid was filtered, washed, dried and finally recrystallized from the proper solvent affording the derivatives 16a-c.

5-[2-Ethoxyquinazolone-3-yl]-2-phenylthiadiazole (16a): Brown crystals from DMF; m.p. 172–173 °C; yield 85%. Anal. for C$_{18}$H$_{14}$N$_4$O$_2$S (m.w. 350); Found: C, 61.88; H, 4.04; N, 16.08; S, 9.17; Calcd: C, 61.71; H, 4.00; N, 16.00; S, 9.14. IR $\nu$ (cm$^{-1}$) 1630 (C=N), 1669 (C=O). MS: m/z (int. %) [M+H]$^+$ 350 (78.0), 352 (31.1), 191 (100), 193 (23.7), 157 (4.4), 159 (0.1), 175 (49.5), 177 (9.7), 162 (38.2), 164 (3.6), 130 (61.2), 132 (4.1), 103 (1.8), 105 (0.2), 78 (0.7), 80 (0.1); $^1$H-NMR (DMSO-d$_6$) $\delta$ 1.22 (t, 3H; -OCH$_2$CH$_3$), $J = 7.1$ Hz), 4.4 (q, 2H; -OCH$_2$CH$_3$), 7.41–7.94 (m, 5H, Ph-H), 7.44–8.20 (4d, 4H, quinazolinone).

5-[2-Ethoxyquinazolone-3-yl]-2-cinnamylthiadiazole (16b): Brown crystals from DMF; m.p. 289-290 °C; yield 85%. Anal. for C$_{20}$H$_{16}$N$_4$O$_2$S (m.w. 376); Found: C, 68.99; H, 4.72; N, 16.00; S, 9.03; Calcd: C, 68.97; H, 4.60; N, 16.09; S, 9.20. IR $\nu$ (cm$^{-1}$) 1633 (C=N), 1689 (C=O). MS: m/z (int. %) [M+H]$^+$ 376 (72.0), 378 (14.9), 191 (100), 193 (17.5), 188 (14.2), 190 (3.1), 175 (48.2), 177 (12.7), 157 (5.7), 159 (0.2), 130 (67.1), 132 (1.8), 129 (0.8), 131 (0.2), 122 (1.3), 124 (0.1), 78 (0.6), 80 (0.1); $^1$H-NMR (DMSO-d$_6$) $\delta$ 1.23 (t, 3H; -OCH$_2$CH$_3$), $J = 7.1$ Hz), 4.43 (q, 2H; -OCH$_2$CH$_3$), 7.42–8.20 (4d, 4H, quinazolinone ArH), 7.35–7.45 (m, 5H, cinnamyl ArH), 7.20, 7.47 (2d, 2H, $J = 15.8$ Hz, olefinic -H).

5-[2-Ethoxyquinazolone-3-yl]-2-phthalamidomethylthiadiazole (16c): Brown crystals from DMF; m.p. 303–304 °C; yield 85%. Anal. for C$_{21}$H$_{15}$N$_5$O$_4$S (m.w. 433); found: C, 58.72; H, 3.66; N, 16.31; S, 7.42; Calcd: C, 58.20; H, 3.46; N, 16.17; S, 7.39. IR $\nu$ (cm$^{-1}$) 1631 (C=N), 1670, 1727, 1776 (3C=O). MS: m/z (int. %) [M+H]$^+$ 433 (58.0), 435 (22.8), 245 (36.4), 247 (3.4), 191 (100), 193, (56.1), 186 (78.0), 188 (12.7), 175 (30.1), 177 (8.1), 157 (3.2), 159 (0.1), 147 (8.3), 149 (0.3), 130 (48.3), 132 (6.4), 122 (4.5), 124 (0.2), 78 (0.3), 80 (0.1); $^1$H-NMR (DMSO-d$_6$) $\delta$ 1.21 (t, 3H; -OCH$_2$CH$_3$), $J = 7.1$ Hz), 4.48 (q, 2H; -OCH$_2$CH$_3$), 5.16 (s, 2H; CH$_2$, phthalimidomethyl), 7.32–7.86 (m, 4H, quinazolinone), 7.94–8.03 (m, 4H, phthalimido).

Competing Interests

The authors declare no conflict of interest.

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*Sample availability:* available from the authors.