Original Article

Ground and surface water for drinking: a laboratory study on genotoxicity using plant tests

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Abstract

Surface waters are increasingly utilized for drinking water because groundwater sources are often polluted. Several monitoring studies have detected the presence of mutagenicity in drinking water, especially from surface sources due to the reaction of natural organic matter with disinfectant. The study aimed to investigate the genotoxic potential of the products of reaction between humic substances, which are naturally present in surface water, and disinfectants using short-term mutagenicity tests.

Significance for public health

The availability of safe drinking water is an important public health problem. Surface waters are increasingly utilized for drinking water because ground-water sources are often polluted by persistent organic pollutants. Several monitoring studies have detected the presence of mutagenic activity in disinfected drinking water, especially from surface sources, due to the reaction of natural organic matter with disinfectant. The choice of the sources as well as the disinfectants for drinking water is one of the most important topics of public health because of chronic exposure to mutagenic compounds. A study model for simulating natural conditions of both deep well and surface water may be a good method for evaluating the genotoxic potential of the products of reaction between humic substances, which are naturally present in surface water, and disinfectants using short-term mutagenicity tests.

Introduction

Surface waters are increasingly utilized for drinking water because groundwater sources are often polluted by persistent organic pollutants. Several monitoring studies have detected the presence of mutagenic activity in disinfected drinking water, especially from surface sources, due to the reaction of natural organic matter with disinfectant. Many studies show that natural organic substances (humic and fulvic acids) present in surface waters may react with disinfectants utilized to potabilize waters to produce numerous disinfection by-products (DBPs) that are potentially harmful to human health.

Numerous DBPs have a mutagenic and/or carcinogenic activity, and a large number of DBPs are able to cause cancer in experimental studies. Furthermore, they may play a role in adverse reproductive outcomes such as inability to conceive, spontaneous abortion and low birth weight. Epidemiological studies in populations using chlorinated drinking water obtained from surface sources have shown some cancer hazards. For this reason, disinfectants such as ozone and chlorine dioxide are used as alternatives to chlorine for water treatment. Humic substances can also react with ozone to produce aldehydes, ketoacids, and carboxylic acids, which contribute to the biodegradable organic carbon content of ozonated water. Yet aldehydes, formed by ozonation of humic substances (such as formaldehyde, acetaldehyde, glyoxal, glyoxylic acid and methylglyoxal), show a clear mutagenic activity.

Humic and fulvic acid concentrations are related to soil and vegetation which are located near water sources, to algae living in water and to seasonal flowering. Humic compounds are amorphous, brown or black, hydrophilic, acidic, polydispersed substances, with very different molecular weights, and constitute the majority of organic materials present in surface water. These substances derive from both living and decayed organisms, from algae living in water and from seasona flowering. Humic compounds are amorphous, brown or black, hydrophilic, acidic, polydispersed substances, with very different molecular weights, and constitute the majority of organic materials present in surface water. These substances derive from both living and decayed organisms, from algae living in water and to seasonal flowering.

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Chlorination of commercial humics was proposed as a model for examination of mutagen formation during water chlorination using Salmonella/microsome test. The increasing use of surface water as a source of drinking water may become a public health problem due to the chronic exposure of an elevated number of people to genotoxic compounds. Water genotoxicity can be detected directly by in vivo tests, using aquatic animals, such as amphipods, fish, newts
or molluscs or plants, e.g., *Zea mays*, *Vicia faba*, *Allium cepa*, and *Tradescantia* spp.

The aim of this study was to investigate the genotoxic potential of the products of reaction between humic substances, which are naturally present in surface waters, and different disinfectants. In order to simulate natural conditions of both deep water (TOC=2.5 mg/L) and surface water (TOC=7.5 mg/L), two different concentrations of commercial humic acids were treated with three different disinfectants: two widely used drinking water disinfectants – sodium hypochlorite (NaClO) and chlorine dioxide (ClO2) – and peracetic acid (PAA, CH3CO-COOH), a disinfectant not yet utilized for drinking water. These humic acid solutions were studied using genotoxicity plant tests: the chromosomal aberration test in *Allium cepa* root cells and in *Allium cepa* and *Vicia faba* root cells (Figure 1).

### Materials and Methods

#### Sample preparation

Stock solutions of humic acid (Sigma-Aldrich Corporation, St. Louis, MO, USA) were prepared as follows: two solutions of humic acid, 12.5 mg/L and 37.5 mg/L in distilled water, were stirred for 1 h at room temperature. Each litre of solution was diluted to 2.5 litres with distilled water and stored overnight at 4°C. These solutions were prepared to obtain 2.5 mg/L and 7.5 mg/L TOC concentrations simulating a ground water and a surface water, respectively (according to Agarwal and Neton, with some modifications).

A solution of commercial sodium hypochlorite (14.65%) used for drinking water treatment was obtained from Solvay Chimica Italia, S.p.A. (Rosignano, LI, Italy). A chlorine dioxide solution (0.23 g/L) was prepared by passing in distilled water a flow of ClO2 produced in a generator provided by Sanipur S.r.l. (Brescia, Italy). A peracetic acid solution containing 15.20% of PAA in equilibrium with hydrogen peroxide (15.20%) was used (Promox S.r.l., Leggiuno, VA, Italy). Humic acid solutions with low and high TOC values were treated with 2.5 and 7.5 mg/L, respectively, of each disinfectant to attain a C:disinfectant molar ratio of 1:1.

The disinfectant residues were monitored in the solutions using colorimetric methods. NaClO concentrations were measured as free dissolved chlorine using a DR 2000 Hach photometer (Hach Company, Loveland, CO, USA) at 530 nm (Hach 1407099 method, adapted from Standard Methods, 1998, CI G 4500). CI O3 was determined by the N,N-diethyl-p-phenylenediamine (DPD) method at 575 nm (Hach 22423/00 method, adapted from Standard Methods, 1998, CI G 4500). For PAA concentrations, the water was treated with catalase, followed by potassium iodide, and the iodine concentration was measured at 530 nm (Hach 14064/99 method, adapted from Standard Methods, 1998, CI G 4500). PAA calibration curves were obtained by comparing the DPD absorbance of known concentrations of PAA. The measures were performed using a Hach DR 2000 photometer after every hour, 10 times, to detect the trend of the disinfection process. Another measure was performed 24 h after the beginning of the treatment. The sample solutions were stored at 4°C until the genotoxicity plant tests were carried out.

Sample solutions were used also for AOX (Adsorbable Organic Halogens) analysis using the HPLC method. AOX value is a measurement used to estimate the total quantity of dissolved halogenated organic material in a water sample. The presence of halogenated organic molecules is indicative of disinfection by-products.

#### Plant genotoxicity tests

**Allium cepa tests**

In a preliminary assay, equal-sized (2.25 cm in diameter) young bulbs of *Allium cepa* were exposed for 72 h in the dark to undiluted, 1:1 and 1:10 solutions, and root length was measured to determine the EC50 (the concentration which gives a 50% reduction in root growth). Root length and other macroscopic parameters (turgescence, change in colour, root tip shape) were used as an index of toxicity. *Allium cepa* Root Anaphase Aberration (AL-RAA) assay and *Allium cepa*/micronucleus (AL/MCN) test were performed using equal-sized young bulbs (2.25 cm in diameter) of *Allium cepa*. The aberration chromosomal test was carried out at concentrations equal to or lower than the EC50 dose. These samples (undiluted, 1:1 and 1:10 diluted) were tested for 6- and 24-h exposures after 72 h of pre-germination in mineral water. After that, the roots were fixed in 1:3 acetic acid-ethanol solution for 24 h and finally stored in 70% ethanol. One thousand cells/root from 5 roots (5000 cells per sample) were scored for mitotic index and micronuclei, 40 anaphases/root from 20 roots (800 anaphasic cells per sample) were scored for anaphase aberrations, such as bridges, laggards and fragments. Mineral water and two humic acid solutions with different TOC concentration were used as negative controls. A positive control was performed using 10 mg/L maleic hydrazide to ensure the effectiveness of the assay.

The analysis of variance and Dunnett’s test were performed for micronuclei, and the chi-square test was performed for data analysis of mitotic index and anaphase aberrations.

#### Tradescantia micronucleus test

The *Tradescantia* micronucleus (Trad/MCN) test was performed using a hybrid of *Tradescantia hirsutiflora* and *Tradescantia subacaulis* (clone #4430). Young inflorescences of *Tradescantia* were directly immersed in the samples (only dilutions 1:1 and 1:10 on the basis of the results obtained in *Allium cepa* preliminary toxicity tests) for 6 h and then maintained in mineral water for a further 24 h of recovery time. After that, they were fixed in 1:3 acetic acid-ethanol solution and stored in 70% ethanol. The buds were then used to prepare slides and the micronucleus frequency in meiotic pollen mother cells was evaluated. Normal tetrads and tetrads containing one or more micronuclei

**Figure 1. Scheme of the research.**
(MCN) were counted from five slides for each experimental group. Over 1500 tetrads were scored for each sample. The data were expressed as MCN/100 tetrads (mean±standard deviation) and analyzed for significance using analysis of variance and Dunnett’s t-test. Negative control was carried out using distilled water. Two humic acid solutions with different TOC concentrations were also used as negative controls. A positive control with 5 mg/L of maleic hydrazide was performed concurrently.

**Vicia faba micronucleus test**

Micronucleus test in *Vicia faba* (*Vicia faba/MCN*) root tips was performed according to standard protocols. After germination of *Vicia faba* seeds in Hoagland’s solution, primary roots were removed for a faster secondary root production, which were then exposed in the dark to the treatment solutions (1:10 dilution in Hoagland’s solution). Two exposure times were studied: a short interval of 6 h, followed by 66 h recovery in fresh Hoagland’s solution, and 72 h exposure, until fixation. At the end of treatment, secondary roots were removed and fixed in 1:3 acetic acid-ethanol mixture. Hoagland’s salt solution and two humic acid solutions with different TOC concentrations were used as negative controls. Positive control treatment was also performed using maleic hydrazide 10–4 M (11.2 mg/L) in Hoagland’s solution (4 h treatment + 44 h recovery).

Feulgen staining of the roots was performed and the cut tips were squashed onto slides. The mitotic index was estimated on 1000 cells/tip and the frequency of micronuclei was determined in 5×104 cells per sample (mean±standard error), analysing 5000 cells per root tip, 2 secondary roots per plant, and 5 plants per experimental point (10 root tips/experimental point, 5×104 cells/point). Statistical analysis of the data was carried out by means of the Mann-Whitney non-parametric test. The Kruskal-Wallis non-parametric ANOVA was also used for control comparison purposes.

### Results

#### Chemical analyses

The results of the chemical analyses for AOX detection carried out on the samples of undiluted humic acids are shown in Figure 2. In the presence of 7.5 mg/L TOC the AOX values are always higher than 2.5 mg/L TOC, particularly for NaClO treatment (220 µg/L). It is noteworthy that the highest AOX values are detected in NaClO and PAA treated humic acid solutions, at both TOC concentrations.

#### Plant genotoxicity tests

**Allium cepa tests**

The results of the preliminary toxicity test in *Allium cepa* carried out on undiluted and diluted (1:1 and 1:10) solutions of humic acid showed very high toxicity in the undiluted solution, therefore the plant genotoxicity tests were performed only at 1:1 and 1:10 diluted humic acid solutions. Exposures to 1:1 and 1:10 humic acid dilutions were carried out for 3, 6 and 24 h.

The results of the *Allium cepa* test on the 1:1 solutions of humic acid are shown in Table 1. Microscopic evaluation of *Allium cepa* roots was not possible due to the presence of toxicity that hides mutagenic activity in bulbs exposed for 24 h to the 1:1 dilution; therefore no data are reported for this experimental point. After 3 h of exposure chromosomal aberrations in *Allium* root cells were higher compared to the negative control for all samples, the untreated humic acid solutions includ-

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**Table 1. Mitotic index, anaphase aberrations and micronuclei in *Allium cepa* root tips exposed for 3 and 6 h to solutions of treated and untreated humic acids diluted 1:1.**

| Samples          | Mitotic index (%) | 3 h exposure time | Anaphase aberrations (%) | Mean±SD | Mitotic index (%) | 6 h exposure time | Anaphase aberrations (%) | Mean±SD |
|------------------|-------------------|-------------------|--------------------------|---------|-------------------|-------------------|--------------------------|---------|
| Negative control | 12.1              | 4.4               | 0.8±0.8                  | 11.0    | 2.8               | 1.0±1.2           |                          |         |
| **TOC 2.5 mg/L** |                   |                   |                          |         |                   |                   |                          |         |
| Humic acids      | 9.9               | 8.3**             | 1.2±1.3                  | 7.6*    | 34.3***           | 2.4±3.4           |                          |         |
| Humic acids+ClO₂ | 9.1               | 7.4***            | 1.0±1.2                  | 8.8     | 49.5***           | 1.6±1.5           |                          |         |
| Humic acids+NaClO| 10.1              | 14.2***           | 0.8±0.8                  | 8.0*    | 57.1***           | 6.2±4.0*          |                          |         |
| Humic acids+PAA  | 9.8               | 10.2***           | 0.2±0.4                  | 8.9     | 68.4***           | 3.2±4.7           |                          |         |
| **TOC 7.5 mg/L** |                   |                   |                          |         |                   |                   |                          |         |
| Humic acids      | 9.5               | 7.2*              | 0.6±0.9                  | 10.3    | 42.5***           | 2.8±0.8*          |                          |         |
| Humic acids+ClO₂ | 9.3               | 7.5**             | 1.2±1.6                  | 9.3     | 42.2***           | 1.4±1.1           |                          |         |
| Humic acids+NaClO| 9.5               | 15.2***           | 1.2±1.1                  | 9.4     | 50.0***           | 1.6±1.1           |                          |         |
| Humic acids+PAA  | 9.8               | 13.8***           | 0.8±1.3                  | 8.1     | 52.0***           | 1.6±1.5           |                          |         |

*P<0.05 statistically significant vs negative control; **P<0.01 statistically significant vs negative control; ***P<0.001 statistically significant vs negative control; °°°P<0.001 statistically significant vs humic acid; positive control: maleic hydrazide (10 mg/L). 1.3% of anaphase aberrations and 7.5±2.1 of micronuclei.
ed. Moreover, NaClO treatment with 2.5 mg/L TOC and NaClO and PAA treatments with 7.5 mg/L TOC induced an increase in mutations compared to humic acid without disinfectant. When using a longer exposure time (6 h), genotoxic damage was much greater but the reduction in mitotic index indicated a strong toxicity, particularly for 2.5 mg/L TOC samples. For this reason the genotoxicity data may be invalidated. On the other hand, no significant MCN increase is detected from any disinfectant-treated humic acid solution, apart from that induced by NaClO treatment, but only for humic acid solution with 2.5 mg/L TOC. In 7.5 mg/L solution a mild increase in MCN was only seen in humic acid solution without disinfectant.

Actually, a consistent MCN increase is the expected consequence after the dramatic mutagenic effect as that evidenced by the anaphase aberration test. Indeed, the clear cytotoxic effect observed in Allium roots after 3 and 6 h exposure in all the tested humic acid solutions did not allow cell cycle progression, thereby preventing the production of micronucleated daughter cells. Therefore the Allium test was repeated after 1:10 dilution of all test solutions.

Table 2 shows the results of the Allium cepa test on the 1:10 dilution of humic acid after 6 and 24 h of exposure. The results showed that all the disinfected samples were positive vs. negative control (undisinfected humic acid solution) after 24 h of exposure: both humic acid solutions (TOC 2.5 mg/L and TOC 7.5 mg/L) treated with ClO₂ induced a significant increase in MCN frequency, whereas NaClO- and PAA-treatments induced anaphase aberrations.

Several samples (disinfected and undisinfected) induced mutations compared to the negative control.

### Tradescantia micronucleus test

The results of the Trad/MCN test are set out in Table 3. Genotoxic effects were found for the solution of humic acid with TOC 2.5 mg/L treated with NaClO and PAA after 6-h exposure to dilutions 1:1 and 1:10, respectively. The solution with higher TOC disinfectated with NaClO induced a very high level of MCN frequency (dilution 1:1). ClO₂ did not induce any genotoxicity in this organism. A slight MCN increase, yet significant at the 1:10 dilution, is registered for untreated humic acid solution with TOC 7.5 mg/L.

### Conclusions

The aim of this research was to study the formation of genotoxic, and potentially carcinogenic, agents in the products of reaction between commercial humic substances and three disinfectants for drinking water, by means of in vivo short-term plant genotoxicity tests, and to compare the effects of two widely used disinfectants, ClO₂ and NaClO, with a new disinfectant, PAA, a potent antimicrobial agent, with many applications in hospitals, laboratories, factories, and wastewater disinfection, but not yet used for drinking water.

All the tested disinfectants induced a clastogenic/aneugenic effect in plant cells, even in the presence of concentrations of organic substances similar to those frequently present in drinking water. All disinfectants determined genotoxic effects in Allium cepa tests. In particular, NaClO- and PAA-treated water samples induced chromosomal aberrations, while ClO₂ induced mainly MCN in Allium cepa. Furthermore, NaClO treatment induced MCN increase in Tradescantia pollen cells and also Vicia faba root cells. Besides, PAA induced MCN in

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**Table 2. Mitotic index, anaphase aberrations and micronuclei in Allium cepa root tips exposed for 6 and 24 h to solutions of treated and untreated humic acids diluted 1:10.**

| Samples         | Mitotic index (%) | 1:10 dilution 6 h exposure (mean±SD) | MCN (mean±SD) | Mitotic index (%) | 1:10 dilution 24 h exposure (mean±SD) | MCN (mean±SD) |
|-----------------|-------------------|--------------------------------------|---------------|-------------------|----------------------------------------|---------------|
| Negative control| 10.6              | 2.1                                  | 0.8±0.8       | 12.1              | 4.7                                    | 0.8±0.8       |
| Humans acids    | 8.8               | 5.4***                               | 0.4±0.5       | 9.5               | 4.7                                    | 1.2±1.6       |
| Humans acids+ClO₂| 10.3              | 2.6                                  | 0.4±0.5       | 10.9              | 4.1                                    | 6.2±1.9***    |
| Humans acids+NaClO| 9.4               | 3.5                                  | 0.4±0.5       | 9.3               | 12.9***                                 | 0.2±0.4       |
| Humans acids+PAA| 9.3               | 4.4***                               | 0.2±0.4       | 7.8***            | 8.9***                                 | 0.6±1.3       |
| TOC 7.5 mg/L    |                   |                                      |               |                   |                                        |               |
| Humans acids    | 10.4              | 3.8*                                 | 2.8±0.8       | 10.7              | 4.0                                    | 1.2±1.6       |
| Humans acids+ClO₂| 11.4              | 4.7**                                | 1.4±1.1       | 7.8***            | 5.1                                    | 6.2±1.9***    |
| Humans acids+NaClO| 8.8               | 4.4*                                 | 1.6±1.1       | 9.8               | 9.6***                                 | 0.2±0.4       |
| Humans acids+PAA| 8.9               | 5.3***                               | 1.6±1.5       | 8.0***            | 10.7***                                | 0.6±1.3       |

*P<0.05 statistically significant vs negative control; **P<0.01 statistically significant vs negative control; ***P<0.001 statistically significant vs negative control; °P<0.05 statistically significant vs humic acid; °°P<0.01 statistically significant vs humic acid; °°°P<0.001 statistically significant vs humic acid; positive control: maleic hydrazide (10 mg/L) 5.9% of anaphase aberrations and 7.5±2.1 of micronuclei.
*Humic acids + NaClO 4.9± 0.96** ° 3.8± 0.71* °

Several studies have shown that treating surface drinking waters with disinfectants such as NaClO, ClO₂, and PAA causes an increase in the mutagenic activity of the treated water, whereas other disinfectant treatments, such as the use of ozone, decreased their mutagenic effects [3,29,35,36].

In our study all the treatments played an important role in DBP formation, but the concentration of organic substances did not appear to influence this; unexpectedly, no relevant differences were found in samples with different values of TOC: highest TOC concentration (7.5 mg/L) did not induce a greater genotoxic effect than 2.5 mg/L TOC concentration samples. Also, low TOC concentration, similar to ground water, may react with disinfectants to produce DBPs with mutagenic effects on plant systems. These observations are partially in accordance with data from Gustavino et al. [37] at least for PAA. Drinking water obtained from surface water sources does not seem to induce a higher increase in mutagenicity than that obtained from ground water.

These experimental conditions suggest that not only surface water rich in organic substances may induce the formation of DBP, and the choice of disinfectant and its dosage are always important.

In the Allium cepa test, exposure time influenced genotoxicity: the longest exposure induced a significant increase in DNA damage in comparison with short exposure. This effect was partially seen in Vicia faba as well. When time exposure was extended, toxicity and mutagenicity effects generally increased, but toxicity probably prevented potential genotoxic activity.

In this study the Allium cepa test was certainly more sensitive to the mutagenic effects produced by disinfected humic acid, but even humic acids alone, without disinfectants, were genotoxic in Allium cepa.

The plant tests seem to be suitable for studying the genotoxicity of commercial humic acids treated with different disinfectants, and the solutions of humic acids were a good model for examining DBP formation during water disinfection, regardless of their concentration. The use of plants may allow the detection of intermediary mutagenic metabolites, which can be produced independently of the organic carbon content.

The use of commercial humic acids in aquatic and terrestrial research has been critically reviewed, however [38]. The authors found large differences in seven samples of commercial humic acids, purchased from different suppliers, in comparison with natural compounds, when they were detected by NMR and IR spectroscopy. Also on the other hand there is much literature in favour of the use of these humic substances as a model for the formation of DBPs. Commercial humic substances for the examination of mutagen formation during water chlorination were used by Li et al. [39] who studied the effect of ClO₂ on DBP formation and found that THMFP could be induced by treatment with mixtures of ClO₂ and Cl₂ in varying ratios. Different disinfectants can contribute to DBP formation in a different way as a function of humic acid composition, structure and functional properties. Moreover, humic acids can exert an antilastogenic action in different plants too [40,42] and this should clarify the behaviour observed in this study at different TOCs, where an increase in TOC did not cause an increase in mutagenicity. The highest TOC probably contributed to greater DBP formation, but at the same time humic acids reduced their genotoxicity. A protective effect against the toxicity of the chlorinated compounds has been reported in in vivo tests by Gustavino et al. [37]. Therefore, the observed mutagenicity is the result of a soft balance in organic matter concentration in water and disinfectant (type, dose and contact time).

Disinfection remains the fundamental treatment of drinking water because the risks to human health from disinfection by-products are extremely small in comparison with the risks associated with inadequate disinfection, but the chronic exposure of a large number of people to mutagenic and/or carcinogetic compounds imposes the adoption of risk-reduction strategies and measures, such as reducing DBP concentrations, changing the disinfection process, using different chemical disinfectants with a lower propensity to produce by-products with the source water, using non-chemical disinfection processes, removing precursor compounds prior to disinfection process and/or removing DBPs prior to distribution. A study model using disinfected humic acid solutions can provide additional information on DBP formation and may help health authorities to evaluate drinking water quality and

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**Table 3. Mean frequency (±SD) of micronuclei in early tetrads of Tradescantia inflorences exposed for 6 h to solutions of treated and untreated humic acids diluted 1:1 and 1:10.**

| Samples | 1:1 dilution (mean ± SD) | 1:10 dilution (mean ± SD) |
|---------|--------------------------|----------------------------|
| Negative control | 5.7±1.3 | 2.9±1.3 |
| TOC 2.5 mg/L | | **°** |
| Humic acids | 6.5±1.7 | 4.7±0.4 |
| Humic acids + ClO₂ | 6.4±2.1 | 5.9±2.0* |
| Humic acids + NaClO | 8.2±1.8 | 8.4±1.9*** ° |
| Humic acids + PAA | 11.3±4.1*** ° | 5.8±3.1 |
| TOC 7.5 mg/L | | **°** |
| Humic acids | 6.6±2.7 | 7.7±4.6* |
| Humic acids + ClO₂ | 6.5±1.9 | 10.9±4.5** |
| Humic acids + NaClO | 20.2±12.5**° | 4.6±1.7 |
| Humic acids + PAA | 11.9±7.8 | 6.0±3.5 |

*P<0.05 statistically significant vs negative control according to Dunnett’s test; **P<0.01 statistically significant vs negative control according to Dunnett’s test; ***P<0.001 statistically significant vs negative control according to Dunnett’s test; positive control: maleic hydrazide (5 mg/L) 28.5±6.

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**Table 4. Mean micronucleus frequencies ±SE (per 5000 cells) in Vicia faba root tips exposed for 6 and 72 h to the humic acid solutions (2.5 and 7.5 TOC) alone or treated with the three disinfectants, after dilution 1:10 in Hoagland solution.**

| Samples | MCN/5000 cells (mean ± SE) |
|---------|---------------------------|
| **6 h exposure time** | **72 h exposure time** |
| TOC 2.5 mg/L | | |
| Humic acids | 2.2±0.39 | 1.7±0.47 |
| Humic acids + ClO₂ | 3.2±0.49* | 2.5±0.67 |
| Humic acids + NaClO | 4.9±0.56**° | 3.8±0.71° |
| Humic acids + PAA | 3.0±0.60 | 2.5±0.62 |
| **TOC 7.5 mg/L** | | |
| Humic acids | 3.2±0.74 | 1.7±0.26 |
| Humic acids + ClO₂ | 3.4±0.43* | 3.5±0.90 |
| Humic acids + NaClO | 5.3±0.61***° | 3.5±0.45**° |
| Humic acids + PAA | 2.8±0.51 | 2.6±0.52 |

*P<0.05 statistically significant vs Hoagland solution according to Mann-Whitney test; **P<0.01 statistically significant vs Hoagland solution according to Mann-Whitney test; ***P<0.001 statistically significant vs 1 mm acid according to Mann-Whitney test; positive control: maleic hydrazide (11.2 mg/L) 28.5±6.
adopt effective strategies for reducing genotoxic compounds in disinfected drinking water.

In conclusion, in agreement with previous studies, NaClO in particular, but also ClO₂ and PAA, was genotoxic in plants and these tests were useful for evaluating the mutagenicity of different drinking water disinfectants using humic acid solutions as a study model.

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