ATP-driven Reduction by Dark-operative Protochlorophyllide Oxidoreductase from "Chlorobium tepidum" Mechanistically Resembles Nitrogenase Catalysis* □

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During chlorophyll and bacteriochlorophyll biosynthesis in gymnosperms, algae, and photosynthetic bacteria, dark-operative protochlorophyllide oxidoreductase (DPOR) reduces ring D of aromatic protochlorophyllide stereospecifically to produce chlorophyllide. We describe the heterologous overproduction of DPOR subunits BchN, BchB, and BchL from "Chlorobium tepidum" in Escherichia coli allowing their purification to apparent homogeneity. The catalytic activity was found to be 3.15 nmol min⁻¹ mg⁻¹ with Kₘ values of 6.1 μM for protochlorophyllide, 13.5 μM for ATP, and 52.7 μM for the reductant dithionite. To identify residues important in DPOR function, 21 enzyme variants were generated by site-directed mutagenesis and investigated for their metal content, spectroscopic features, and catalytic activity. Two cysteine residues (Cys⁹⁷ and Cys¹³¹) of homodimeric Bchl₂ are found to coordinate an intersubunit [4Fe-4S] cluster, essential for low potential electron transfer to (BchN)₂ as part of the reduction of the protochlorophyllide substrate. Similarly, Lys¹⁶⁰ and Leu¹⁷⁶ are crucial to ATP-driven electron transfer from BchL₁. The activation energy of DPOR electron transfer is 22.2 kJ mol⁻¹ indicating a requirement for 4 ATP per catalytic cycle. At the amino acid level, Bchl is 33% identical to the nitrogenase subunit NiFH allowing a first tentative structural model to be proposed. In (BchN)₂, we find that four cysteine residues, three from BchN (Cys²¹, Cys⁶⁶, and Cys¹⁰³) and one from BchB (Cys³⁴⁶), coordinate a second intersubunit [4Fe-4S] cluster required for catalysis. No evidence for any type of molybdinum-containing cofactor was found, indicating that the DPOR subunit BchN clearly differs from the homologous nitrogenase subunit NiFD. Based on the available data we propose an enzymatic mechanism of DPOR.

Protochlorophyllide (Pchlide)² is a central metabolite for the biosynthesis of chlorophylls (Chl) and bacteriochlorophylls (bChl). In photosynthetic organisms two distinct enzymes catalyze the stereospecific reduction of ring D of the aromatic Pchlide to form chlorophyllide (Chlide) (1–3) (Fig. 1). The first enzyme is the light-dependent Pchlide oxidoreductase (LPOR; NADPH Pchlide oxidoreductase, EC 1.3.1.33). To achieve the chemically difficult reduction of the aromatic ring system, the complex of monomeric LPOR and substrate Pchlide must absorb light energy prior to NADPH-dependent reduction (4–6) making LPOR a dominant player in the light-dependent greening of flowering plants (angiosperms) (7, 8). The second Pchlide-reducing enzyme is the light-independent (dark) Pchlide oxidoreductase (DPOR), which shares no amino acid sequence homology to the LPOR system. It consumes ATP to drive the production of Chlide. Some anoxygenic bacteria only encode DPOR (9), whereas angiosperms exclusively produce LPOR. Cyanobacteria, algae, and gymnosperms possess both LPOR and DPOR (1). DPOR enables these organisms to synthesize bChl or Chl in the dark.

In bChl-synthesizing organisms the genes bchN, bchB, and bchl encode the three subunits of DPOR. In Chl-synthesizing organisms the corresponding genes are known as chlN, chlB, and chlL (10–12). Protein subunits of DPOR share significant amino acid sequence similarity with NiFD, NiFK, and NiFH, the three subunits of nitrogenase. Among the subunits of DPOR, Bchl is most similar to its counterpart NiFH sharing 33% amino acid sequence identity (13). The remaining two DPOR subunits BchN and BchB are similarly related to NiFD and NiFK, although sequence identities at ~15% (for both comparisons) are not as pronounced (14).

The well characterized nitrogenase protein complex consists of homodimeric Fe-dinitrogen reductase (Fe-protein, NiH₂) and tetrameric MoFe-protein complex (NiFd/NiK₂) responsible for the reduction of dinitrogen to ammonia (15–17). NiFH₂ binds an intersubunit [4Fe-4S] redox cluster symmetrically coordinated by two cysteine residues (Cys⁹⁷ and Cys¹³², Azotobacter vinelandii numbering) from each NiFH subunit. This redox center is required for the ATP-driven chlorophyllide oxidoreductase.

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2 The abbreviations used are: Pchlide, protochlorophyllide; bChl, bacteriochlorophyll; Chl, chlorophyll; Chlide, chlorophyllide; GST, glutathione S-transferase; ICP-MS, inductively coupled plasma mass spectrometry; HPLC, high pressure liquid chromatography; LPOR, light-dependent protochlorophyllide oxidoreductase; DPOR, dark-operative protochlorophyllide oxidoreductase.
DPOR from C. tepidum, Chlorophyll Biosynthesis, and Nitrogenase

transferring electrons from ferredoxin to (NifD/NifK)_2. These two cysteines are perfectly conserved in all Bchl/Chl protein sequences (10).

The tetrameric (NifD/NifK)_2 MoFe complex bears two types of metallocenters, an [8Fe-7S] cluster (P-cluster) located at the interface of the NifD and NifK subunits and a [1Mo-7Fe-9S-homocitrate] or MoFe cofactor located within each NifD subunit (Fig. 2). The P-cluster is thought to mediate electron transfer from NifH_R to the MoFe cofactor, where final dinitrogen reduction takes place (18).

BchN and BchB together contain only four highly conserved cysteines, compared with six cysteines coordinating the [8Fe-7S] P-cluster in NifD/NifK. Three are found in BchN (Cys^{21}, Cys^{46}, Cys^{103}, C. tepidum numbering) and one in BchB (Cys^{95}). A bridging Fe-S center, although presumably smaller than the P-cluster, may thus be located at the interface of BchN and BchB. Residues involved in MoFe cofactor binding in NifD are not conserved in BchN implying that this part of dinitrogen reduction does not have a counterpart in reduction of Pchl.

Several protein complexes of dinitrogenase reductase and the MoFe-protein have been trapped using AlF_4 or BeF_3 together with Mg-ADP, or by deleting Leu^{127} in the ATP-binding motif of NifH (19). The [4Fe-4S] cluster of NifH_R has, moreover, been shown to be a better reductant when in complex with the MoFe-protein, than in its uncomplexed form (20). These findings indicate that the Bchl subunits of DPOR may similarly catalyze an ATP-driven transfer of electrons from a bound [Fe-S] cluster to catalytic protein complex BchNB containing a second Fe-S redox center. Interestingly, ATP-dependent electron transfer is not restricted to dinitrogenase. Instead, Fe-dinitrogenase reductase-like subunits also occur in enzymes such as 2-hydroxyacyl-CoA dehydratase, referred to as “archerases” due to their ability to “activate single electrons” by an ATP-induced change in the redox potential of a [4Fe-4S] center (21).

DPOR from Rhodobacter capsulatus, studied using a homologous expression system, consists of Bchl and a BchNB complex. Its enzymatic activity requires both ATP and a reducing agent such as dithionite (22, 23), whereas ferredoxin is its physiological electron donor (22). Because R. capsulatus DPOR activity declines rapidly after affinity purification, crude cellular extracts were used in these experiments (22).

We have established the first heterologous milligram range production and purification system for Chlorobium tepidum DPOR in Escherichia coli, allowing the kinetic, spectroscopic, and structural characterization of recombinant wild-type and mutant BchNB and Bchl including two [4Fe-4S] clusters and their coordinating cysteine residues.

**EXPERIMENTAL PROCEDURES**

Production and Purification of C. tepidum DPOR in E. coli—The complete bchNBL operon from C. tepidum (strain ATCC 49652) was PCR-amplified using primers CAGGATCCATGAGTTTAGTATTGGCC and cloned into BamHI and NcoI sites of pGEM-6Ril to generate pGEX-6Ril. The bchL gene was analogously amplified using primers CAGGATCCATGATTTAGTATTGGCC and CAATGCGGCCGCTTACTGCCAGCCACC and cloned into BamHI and NcoI sites of pGEX-6Ril to yield pGEX-bchNL. The plasmids were transformed into E. coli BL21(DE3) Codon Plus RIL to produce the BchNLB complex or only Bchl. Cells were cultivated at 17 °C in 500 ml of LB medium with 1 mM Fe(III)-citrate. Protein production was induced by 50 μM isopropyl-β-thiogalactoside at 0.6 OD_{578}. After 16 h of cultivation 1.7 mM dithionite was added and cultivation was continued without agitation for 2 h at 18 °C in an anaerobic chamber (Coy Laboratories, Grass Lake, MI) to allow [Fe-S] cluster formation. All remaining steps were performed under anaerobic conditions at oxygen partial pressure below 1 ppm (Oxygen detector, Coy Laboratories, Grass Lake, MI). Solutions were N_2-saturated prior use. Cells were centrifuged, resuspended in 15 ml of glutathione-Sepharose (GE Healthcare) equilibrated with lysis buffer (100 mM HEPES-NaOH (pH 7.5), 10 mM MgCl_2, 1 M glycerol, 10 mM dithiothreitol), and disrupted by a single passage through a French press at 1500 p.s.i. into an anaerobic bottle. Following 60 min of centrifugation at 175,000 g at 4 °C, the supernatant was added to 500 μl of glutathione-Sepharose (GE Healthcare) equilibrated with lysis buffer. After washing with 20 ml of phosphate-buffered saline containing 10 mM dithiothreitol (wash buffer) the recombinant fusion protein GST-BchN in complex with BchB or GST-Bchl alone were eluted using 2 ml of 50 μM Tris-HCl (pH 8.0) containing 10 mM reduced glutathione. Fractions containing the BchNB complex or GST-Bchl were identified by SDS-PAGE.
Purification of the Ternary BchNB Complex—To obtain a stable ternary DPOR complex, all steps of the affinity purifications were carried out in the presence of 2 or 10 mM BeF$_3$, or 100 or 200 µM AlF$_3$ and 5 mM Mg-ADP. Cell-free extracts were incubated with metallofluorides and Mg-ADP 30 min prior to affinity chromatography.

Site-directed Mutagenesis of DPOR Genes and Ribosomal Binding Sites—Up to four nucleotides of the plasmid pGEX-bchNBL were exchanged using the QuikChange™ site-directed mutagenesis kit (Stratagene). Oligonucleotides to incrementally optimize the ribosomal binding site of bchN were exchanged using the site-directed mutagenesis kit (Stratagene). Oligonucleotides to incrementally optimize the ribosomal binding site of bchN were exchanged using the QuikChange™ site-directed mutagenesis kit (Stratagene). Oligonucleotides to incrementally optimize the ribosomal binding site of bchN were exchanged using the QuikChange™ site-directed mutagenesis kit (Stratagene). Oligonucleotides to incrementally optimize the ribosomal binding site of bchN were exchanged using the QuikChange™ site-directed mutagenesis kit (Stratagene). Oligonucleotides to incrementally optimize the ribosomal binding site of bchN were exchanged using the QuikChange™ site-directed mutagenesis kit (Stratagene). Oligonucleotides to incrementally optimize the ribosomal binding site of bchN were exchanged using the QuikChange™ site-directed mutagenesis kit (Stratagene).

Point substitutions in BchN were introduced using the oligonucleotides: C8S, GTTCAAGCAGAT; C215S, CCACAGCTTCAGGCGGTCCCTG; C255S, CGCGGTCGACGTGCGTTGCC; C465S, GTGATCCACCAAGGCGGCACGCTTCC; C1035S, CCGTGCTTTTTCAGGTACCTCAGGCTCTC; C1535S, GTGGCTTTTACTCAGGTACCTCAGGCTCTC; C2205S, CTTCGCTGCTGGTCCGGCCCTG; and C3095S, GGAAGTTGTCAGAGAGAGCCACGG.

Oligonucleotides for the exchange of codons in bchN were: C94S, GTAGCCCCGAGT; C153S, GTGCCCTTCGCTG; C465S, GGTACCCAAGGCGGCACGCTTCC; C1035S, CCGTGCTTTTTCAGGTACCTCAGGCTCTC; C1535S, GTGGCTTTTACTCAGGTACCTCAGGCTCTC; C2205S, CTTCGCTGCTGGTCCGGCCCTG; and C3095S, GGAAGTTGTCAGAGAGAGCCACGG.

Oligonucleotides for the exchange of codons in bchL were: C94S, GACGAAAAG; C153S, GTGCCCTTCGCTG; C465S, GGTACCCAAGGCGGCACGCTTCC; C1035S, CCGTGCTTTTTCAGGTACCTCAGGCTCTC; C1535S, GTGGCTTTTACTCAGGTACCTCAGGCTCTC; C2205S, CTTCGCTGCTGGTCCGGCCCTG; and C3095S, GGAAGTTGTCAGAGAGAGCCACGG.

Oligonucleotides for the exchange of codons in bchL were: C94S, GACGAAAAG; C153S, GTGCCCTTCGCTG; C465S, GGTACCCAAGGCGGCACGCTTCC; C1035S, CCGTGCTTTTTCAGGTACCTCAGGCTCTC; C1535S, GTGGCTTTTACTCAGGTACCTCAGGCTCTC; C2205S, CTTCGCTGCTGGTCCGGCCCTG; and C3095S, GGAAGTTGTCAGAGAGAGCCACGG.

Protein Concentrations—Concentrations of GST fusion proteins in crude cellular extracts were determined using the GST-detection module (GE Healthcare). The concentration of purified proteins was determined employing the BCA (bicinchoninic acid) protein assay kit (Pierce) and bovine serum albumin as a standard.

N-terminal Amino Acid Sequence Determination—Edman degradation was used to confirm the identity of purified proteins and quantify protein subunits from Western blots.

UV-Visible Light Absorption Spectroscopy—UV-visible light spectra of purified recombinant BchNB complexes and BchL were recorded using a V-550 spectrophotometer (Jasco, Groß Umstadt, Germany) under strict anaerobic conditions.

Iron Determination Method—Protein-bound iron was determined colorimetrically with o-phenanthroline after acid denaturation of purified BchNB or BchL (24). The iron content of wild-type BchNB and BchL was confirmed by commercial inductively coupled plasma mass spectrometry (ICP-MS) (CURRENTA Bayer-Analytics, Leverkusen, Germany).

Detection of Potential Flavins—Standard detection methods for FMN and FAD cofactors were performed as described elsewhere (25, 26).

Analysis of a Potential Heme Cofactor—The presence of a potential heme cofactor was screened as described elsewhere (27).

Pchlide Preparation—Pchlide was isolated from the bchL-deficient R. capsulatus strain ZY5 as described earlier (23).

Pchlide Reduction Assay—DPOR activity was measured in 250-µl assays in 100 mM Hepes-NaOH (pH 7.5), 2 mM ATP, 5 mM MgCl$_2$, 20 µM Pchlide, and 10 mM dithiothreitol as electron donor and an ATP regenerating system (20 mM creatine phosphate and 20 units of creatine phosphokinase/assay). This standard DPOR assay was supplemented with 30–100 µL of cell-free E. coli extract containing overproduced GST-BchN, BchB, and BchL. In DPOR reconstitution assays, 20 µg of BchNB complex and 20–60 µg of GST-BchL fusion protein were added. After 30–70 min at 35 °C under strict anaerobic conditions in the dark, reactions were stopped by 1 mL of acetone. Pchlide and Chlide in supernatants obtained by centrifugation at 10 min at 12,000 x g were spectrophotometrically quantified using extinction coefficient ε$_{626}$ = 30.4 mM$^{-1}$ cm$^{-1}$ for Pchlide (23) and ε$_{665}$ = 74.9 mM$^{-1}$ cm$^{-1}$ for Chlide (28). To investigate the ability of ferredoxin to function as electron donor, the reconstitution assay was supplemented with a ferredoxin regenerating system (13 mM glucose 6-phosphate, 1.1 unit of glucose-6-phosphate dehydrogenase from Torula yeast (Sigma), 1.64 mM NADP$^+$, and 0.025 units of ferredoxin-NADP$^+$ oxidoreductase from spinach (Sigma)). The assay was also supplemented with commercially available ferredoxins from Clostridium, Porphyra, or spinach (Sigma) or with recombinantly produced ferredoxin from Synechococcus (29) at a concentration of 25 µg/mL, respectively.

Activation Energy of DPOR Catalysis—To determine the activation energy of ATP-driven DPOR catalysis, the standard assay was repeated between 291 and 319 K. The initial rate of Chlide formation was used to calculate the rate constant at each temperature and the activation energy was determined by the Arrhenius equation,

$$\ln k = \left( \frac{-E_a}{R \cdot T} \right) + c$$  
(Eq. 1)

where $k$ is the rate constant, $E_a$, the activation energy, $R$ the gas constant (8.314 kJ mol$^{-1}$ K$^{-1}$), $T$ the temperature (K), and $c$ a reaction constant.

CN$^−$ Inhibition of DPOR—To analyze the inhibitory effect of CN$^−$ on DPOR catalysis, standard DPOR assays were carried out in the presence of 0–60 mM sodium cyanide.

Molecular Mass Determination—Analytical gel permeation chromatography was performed using a Superdex 200 HR 10/30 column (GE Healthcare), equilibrated with wash buffer containing 100 mM NaCl. The column was calibrated with protein standards (Molecular Weight Marker Kit MW-GF 1000, Sigma) at a flow rate of 0.5 ml min$^{-1}$. A 15-µl sample of purified BchNB or BchL (~45 µg) was run under identical conditions (30).
Molybdadenum Cofactor Analysis—To identify a possible molybdadenum cofactor coordinated by BchN, the nit-1 reconstitutions assay supplemented with 4 \( \mu \)g of BchNB was used (31). 1.2 mg of BchNB were used to detect the stable oxidation product FormA of the cofactor (32). Subsequent product dephosphorylation, sample preparation, and HPLC analysis were performed as published (33).

DPOR Production by Strains of E. coli Defective in Molybdadenum Uptake—DPOR was produced by transforming pGEX-bchNBL* into E. coli RK-5207, a strain unable to import molybdadenum from the medium (34, 35). To overcome the phenotype of the RK-5207 mutant, DPOR was concurrently produced in the presence of 1 and 10 mM Na₂MoO₄ in LB medium. DPOR activity was tested in cell-free extracts.

Molybdadenum and Vanadium Determination by ICP-MS—To rule out the involvement of molybdadenum or vanadium in DPOR catalysis, ICP-MS was performed (36, 37). Samples containing 5 \( \mu \)g of purified protein were applied to the ICP-MS.

RESULTS AND DISCUSSION

C. tepidum DPOR Produced in E. coli Is Enzymatically Active—The complete bchNBL operon of the thermophilic green sulfur bacterium C. tepidum was cloned into a standard E. coli expression vector, producing the DPOR subunit BchN as an N-terminal glutathione S-transferase fusion protein (pGEX-bchNBL). However, amounts of recombinant proteins and DPOR activity were found to be low. Replacing the C. tepidum-type ribosomal binding site upstream of bchB (CCACCGAACCA) and bchL (CCTCTATCATAC) with an E. coli type ribosomal binding site (GGAAACGATGATT) resulted in an expression vector (pGEX-bchNBL*) that allows significant amounts of BchB and BchL to be produced (data not shown). Concomitantly, the DPOR activity of E. coli cell extracts increases from 0.29 to 3.15 nmol min⁻¹ mg⁻¹ (standard DPOR assay, 35 °C) (Fig. 3B, traces a–d), a significant improvement in specific activity compared with ∼40 pmol min⁻¹ mg⁻¹ for DPOR from R. capsulatus cell-free extracts (23). Despite distinct homology to nitrogenase, DPOR activity does not require further enzymes, such as molybdadenum cofactor biosynthesizing or incorporating proteins, known to be essential to nitrogenase-producing organisms. Orthologues of genes encoding these proteins are not found in the genome of the E. coli production host.

Purification of the Heterotetrameric BchNB Complex—DPOR was maintained under strict anaerobic conditions throughout, including during cell harvesting, cell disruption, protein purification, and activity assays. BchN was purified from E. coli pGEX-bchNBL* extracts. GST-BchN fusion protein was coupled to glutathione-Sepharose, washed extensively, and either eluted intact with 10 mM glutathione (Fig. 3A, lane 2)
The BchNB complex was estimated to be in equimolar amounts. The molecular mass of BchL2 was identified in equimolar amounts. The molecular mass of BchNB and BchL. 45

3

Overall, 1.5 mg of pure BchL2 were purified from

B

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equimolar amounts. The molecular mass of

about 60,000 (Fig. 4

M

280 and 428 nm. Protein standards were bovine serum albumin (BSA) (M = 66,000), yeast alcohol dehydrogenase (M = 150,000), β-amyrase (M = 200,000), and apoferritin (M = 443,000). The inferred mass of BchNB is M = 210,000 ± 10,000 indicating a BchNB tetramer, whereas that of BchL is M = 60,000 ± 5,000 indicating a BchL dimer (BchL2).

Purification of the Homodimeric BchL Complex—To allow reconstitution of DPOR, the subunit BchL was separately purified from E. coli extracts. The GST-BchL fusion protein was purified by affinity chromatography and eluted as GST fusion protein with 10 mM glutathione buffer or cleaved off GST by PreScission protease. SDS-PAGE analysis indicates a single protein of M = 57,000 for GST-BchL (Fig. 3A, lane 4) or M = 30,000 for BchL. Analytical gel filtration revealed a relative molecular mass of about 60,000 (Fig. 4C) for BchL indicating that purified BchL from C. tepidum is homodimeric, henceforth denoted BchL2. Overall, 1.5 mg of pure BchL2 were recovered from 5 liters of E. coli culture.

Reconstitution of DPOR Activity—To reconstitute DPOR activity in vitro, 20 μg (0.1 nmol) of (BchNB)_2 and 20–60 μg (0.7–2 nmol) of BchL were assayed using 20 μM Pchlide in the presence of 10 mM dithionite as reductant, 2 mM ATP, and an ATP regenerating system. After 20 min incubation under anaerobic conditions at 35 °C, the reactions were stopped by adding 80% (v/v) acetone. Photometric analysis indicated that the reconstituted C. tepidum enzyme efficiently converts Pchlide (absorbance maximum 626 nm) into Chlide (absorbance maximum 665 nm; Fig. 3B, trace f). Chlide production was not detected if BchNB (trace f), BchL (trace g), dithionite, or the ATP regenerating system were omitted, or when the assay was performed under aerobic conditions (data not shown).

Kinetic Properties of DPOR Catalysis—To establish the K_m values for C. tepidum DPOR we determined the initial velocity of Chlide formation over a broad range of Pchlide, ATP, and dithionite concentrations using the outlined DPOR assay and saturating concentrations for the residual substrates. For Pchlide, ATP, and the artificial co-substrate dithionite, Chlide formation followed Michaelis-Menten kinetics. The apparent K_m for Pchlide was found to be 6.1 μM, slightly lower than the 10.6 μM measured for DPOR from R. capsulatus (22).

Ferredoxin-dependent C. tepidum DPOR Activity—To identify the true electron donor of DPOR, we replaced dithionite with [2Fe-2S]-ferredoxins from Synecochoccus sp. PCC 7002, Porphyra umbilicalis, Spinacia oleracea, and with a [4Fe-4S]-ferredoxin from Clostridium pasteurianum in standard DPOR assays, supplemented with a ferredoxin-regenerating system: glucose 6-phosphate, glucose-6-phosphate dehydrogenase from Torula yeast (Sigma), NADP^+, and ferredoxin-NADP^+ oxidoreductase from spinach (Sigma). Only the ferredoxin from Synecochoccus achieved 10% of the dithionite-supported DPOR activity. As other ferredoxins fail to sustain DPOR activity, we propose that a plant-type [2Fe-2S]-ferredoxin might be the electron donor for C. tepidum DPOR in vivo, corroborating the observation that [2Fe-2S]-ferredoxin from maize is an electron donor for DPOR from R. capsulatus (22).

Thermodynamic Properties of Pchlide Reduction by C. tepidum DPOR—Initial rates of Chlide formation were used to calculate rate constants at various temperatures. An Arrhenius plot of ln k versus 1/T indicates an activation energy for DPOR electron transfer of 22.2 kJ mol^-1 (Fig. 3C), only slightly higher than the 18.8 kJ mol^-1 reported for light-driven Pchlide reduction by LPOR (6) but dramatically lower than the 90–97 kJ mol^-1 required for the electron transfer from the Fe-protein to the MoFe-protein of nitrogenase (38, 39). The nitrogenase reaction is exergonic by ~64 kJ mol^-1 (40). Overall, nitrogenase requires 16 ATP molecules to overcome the kinetic barrier for the reduction of nitrogen (40). Pchlide reduction is similarly
exergonic with a ΔH = −40 kJ mol⁻¹ (5). As the kinetic barrier to DPOR catalysis is only a quarter that of nitrogenase, at least 4 molecules of ATP would be required to transfer two electrons during the reduction of Pchlide. The even number of ATP molecules would match the dimeric nature of BchL2, whereas electron transfer presumably occurs via the inter-subunit [Fe-S] cluster as in nitrogenase.

CN⁻ Resistance of DPOR—To establish whether CN⁻ also uncouples electron transfer from electron transport in DPOR as observed in nitrogenase (41), the effect of CN⁻ on DPOR catalysis was analyzed with the standard DPOR assay. Plotting the DPOR activity (Chlide formation) against CN⁻ concentration (Fig. 3D) indicates that 36 mM NaCN is required to reduce DPOR activity by half, significantly higher than the 5 mM NaCN for A. vinelandii nitrogenase (Mo-nitrogenase) (42). The metal centers of (BchNB)₂ are thus clearly not as sensitive with respect to CN⁻ as those of nitrogenase, indicating distinct differences in the metal centers of DPOR and nitrogenase.

Cysteine Residues Cys⁹⁷ and Cys¹³² of BchL Coordinate an Intersubunit Iron-Sulfur Cluster—Purified, GST-tagged BchL2 is brownish in color. A UV-visible light absorption spectrum recorded under strict anaerobic conditions reveals an absorption peak at 428 nm typical for a [4Fe-4S] cluster (Fig. 4B, spectrum a). The absorption in the range of 390–428 nm bleaches after addition of the reductant dithionite (2 mM, 1 h) (Fig. 4B, spectrum b), whereas oxidation by oxygen exposure for 15 h (Fig. 4B, spectrum c) as well as EDTA treatment (10 mM, 2 h) completely abolishes the absorption in this area of the spectrum. This observation is in accordance to UV-visible spectra obtained for proteins containing iron-sulfur redox clusters (43–45). By applying an extinction coefficient for [4Fe-4S] clusters of ε₄²₈ = 15 mM⁻¹ cm⁻¹ (44, 46) we calculated 0.9 mol of [4Fe-4S] cluster per mol of BchL₂. This ratio was confirmed by ICP-MS quantification revealing 3.62 mol of iron/mol of BchL₂.

Amino acid sequence alignments of BchL proteins reveal five conserved cysteines. Each cysteine was individually substituted by serine to identify potential iron-sulfur cluster ligands. Only variants C97S and C131S cause a significant loss in DPOR activity to 0.5%, whereas C38S and C162S retain 100% and C239S 80% of wild-type activity. Correspondingly, the iron and sulfur content of BchL variants C97S and C131S only were reduced (Table 1). These cysteines thus presumably coordinate the inter-subunit [4Fe-4S] cluster, analogous to Cys⁹⁷ and Cys¹³² in NifH₂. These results are in good agreement with a recent study for R. capsulatus DPOR, where an EPR signal for a [4Fe4S] cluster was obtained (47).

Residues Lys¹⁰ and Leu¹²⁶ of BchL Are Essential to Catalysis—In nitrogenase, binding of NifH₃ to (NifDK)₂ is dependent upon ATP-induced conformational changes. Subsequent ATP hydrolysis induces further conformational rearrangements that decrease the redox potential of the NifH₃ [4Fe-4S] cluster, transferring a single electron to (NifDK)₂, followed by the dis-

| mutation | [Fe-S] coordinating Cys | residues conserved in nitrogenase | enzymatic turnover [%] | iron content [%] | sulfur content [%] |
|----------|-------------------------|----------------------------------|------------------------|-----------------|-------------------|
| bchN C8S | +                       | -                                | 100                    | 100             | 100               |
| bchN C21S | +                      | +                                | 100                    | 23              | 100               |
| bchN C25S | +                      | -                                | 100                    | 100             | 100               |
| bchN C46S | +                      | +                                | 100                    | <0.5           | 100               |
| bchN C103S | +                      | +                                | 100                    | 100             | 100               |
| bchN C153S | +                      | -                                | 100                    | 100             | 100               |
| bchN C220S | +                      | +                                | 100                    | 100             | 100               |
| bchN C309S | +                      | +                                | 100                    | 100             | 100               |
| bchL C94S | +                      | -                                | 100                    | 100             | 100               |
| bchL C94A | +                      | -                                | 100                    | 100             | 100               |
| bchL C217S | +                      | -                                | 100                    | 100             | 100               |
| bchL C373S | +                      | +                                | 100                    | <0.5           | 100               |
| bchL C382S | +                      | +                                | 100                    | 83              | 100               |
| bchL C389S | +                      | +                                | 70                     | 100             | 100               |
| bchL C38S | +                      | +                                | 100                    | 100             | 75               |
| bchL C97S | +                      | +                                | 100                    | <0.5           | 100               |
| bchL C131S | +                      | +                                | 100                    | <0.5           | 100               |
| bchL C162S | +                      | +                                | 100                    | 100             | 66               |
| bchL C239S | +                      | +                                | 80                     | 55              | 52                |
| bchL K10A | +                      | -                                | 100                    | <0.5           | 100               |
| bchL L126 | +                      | +                                | 100                    | <0.5           | 100               |

Table 1: Enzymatic activities, iron, and sulfur content of mutant DPOR enzymes

The specific enzymatic activity of the wild-type DPOR (3.15 nmol min⁻¹ mg⁻¹) was set as 100%. The iron and sulfur contents per mol of purified wild-type (BchNB)₂ and BchL₂ were set as 100%. DPOR protein variants bchN C46S, bchN C103S, bchL C97S, bchL C131S, bchL K10A, and bchL L126 were additionally tested with 260 μg of (BchNB)₂ and 200 μg of BchL₂, per assay with an increased incubation time for up to 4 h. Obtained activities were below to the detection limit of the assay at 15.75 nmol min⁻¹ mg⁻¹.
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FIGURE 5. Ribbon diagram of dimeric BchL. NiFH2 in the ADP-AlF4– bound state (PDB code 1M34) was used to generate the model of BchL2. Dimeric BchL is shown in two mutually perpendicular orientations. The [4Fe-4S] cluster is coordinated by cysteines Cys97 and Cys131 of each subunit at the interface of the dimer. The positions of Lys10 (P-loop), Leu126 (switch II), and the ATP binding site are shown.

The conservation of residues at the dimer interface, the iron-sulfur cluster, and the ATP-binding site, supports the idea of a bridging [4Fe-4S] cluster in BchL2 (see above) coordinated by cysteine residues Cys97 and Cys131 and that BchL2 utilizes a related ATP-controlled gating mechanism to transfer electrons to (BchNB)2. The structural BchL2 model, furthermore, places residues Cys38, Cys162, and Cys239 within the hydrophobic core of each subunit, corroborating the interpretation that these cysteines are not involved in iron-sulfur cluster coordination. Substituting Cys239 by serine presumably destabilizes the protein as a whole, explaining the observed partial loss of DPOR activity of the mutant C239S.

DPOR Catalysis Is Independent of a Molybdenum Cofactor—The presumed evolutionary relationship of DPOR to nitrogenase is supported by the similarity of the amino acid sequences of BchN, BchB, and BchL to those of NifD, NifK, and NifH as well as by the identical oligomeric state of (BchNB)2 and (NifDK)2, and of BchL2 and NiFH2. This raises the question regarding conserved metal centers.

To identify a potential molybdopterin cofactor, (BchNB)2 was analyzed by the nit-1 complementation assay. The mutant nit-1 of *Neurospora crassa* is deficient in molybdenum cofactor biosynthesis, producing a stable apo-form of nitrite reductase. In cell-free extracts of this mutant, nitrite reductase activity can be restored by adding molybdenum cofactor (31). Purified (BchNB)2 is not able to complement this mutant. HPLC analyses failed to detect the stable oxidation product FormA-dephospho of molybdopterin in purified (BchNB)2. Next we produced DPOR in *E. coli* strain RK-5207 deficient in molybdenum uptake (52). Mo-protein synthesis in RK-5207 may be rescued by adding MoO42– to the growth medium (34, 35). The amount of active DPOR produced by *E. coli* RK-5207 is similar to that of *E. coli* strain DH10B without addition of MoO42–. Instead 1 mM MoO42– in
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the growth medium decreases DPOR activity by ~70%. Clearly, production of active DPOR in E. coli does not require molybdenum.

Correspondingly, ICP-MS analyses of purified (BchNB)$_2$ indicated molybdenum and vanadium concentrations no higher than those of the negative controls. (BchNB)$_2$, in contrast to the MoFe-protein of nitrogenase, thus does not bind nor require a molybdenum cofactor.

(BchNB)$_2$ Contains Two Iron-Sulfur Clusters—Under anaerobic conditions, concentrated (BchNB)$_2$ solutions are brownish in color. UV-visible light absorption spectra correspondingly revealed an absorption maximum at 428 nm typical for [4Fe-4S] clusters (Fig. 4a, spectrum a). Reduction with dithionite (2 mM, 1 h) decreases the absorption maximum (Fig. 4a, spectrum b), whereas oxygen exposure for 15 h as well as EDTA treatment (10 mM, 2 h) completely bleaches the ligands. BchN variants C21S, C46S, and C103S show serines by site-directed mutagenesis to identify [Fe-S] cluster BchN (eight) and BchB (five) were individually replaced by corresponding C-terminal regions (incorporating Lys10) and the switch II region might be the natural electron donor of DPOR. The P-loop (Leu126) are crucial for ATP-binding and dynamic positioning of C46SBchN, C153SBchN, C220SBchN, C309SBchN and C217SBchB, respectively.

EDTA treatment (10 mM, 2 h) completely bleaches the absorption in the range of 390–428 nm as described elsewhere (43–45). We calculated 1.8 mol of [4Fe-4S] cluster/mol of (BchNB)$_2$ when using a standard extinction coefficient for [4Fe-4S] clusters of $e_{410} = 15$ M$^{-1}$ cm$^{-1}$ (44, 46). ICP-MS analyses result in 7.33 mol of iron/mol of (BchNB)$_2$ indicating the presence of two [4Fe-4S] clusters per (BchNB)$_2$ tetramer. Standard methods failed to detect additional heme and flavin cofactors.

Cysteines Cys$^{21}$, Cys$^{46}$, Cys$^{103}$ of BchN and Cys$^{94}$ of BchB Participate in Iron-Sulfur Cluster Coordination—All cysteines in BchN (eight) and BchB (five) were individually replaced by serines by site-directed mutagenesis to identify [Fe-S] cluster ligands. BchN variants C21S, C46S, and C103S show <0.5% residual activity and are essentially inactive (Table 1). BchB variants C94S and C389S, by contrast retain 20 and 70% of the residual activity and are essentially inactive (Table 1). BchB region (incorporating Lys10) and the switch II region might be the natural electron donor of DPOR. The P-loop (Leu126) are crucial for ATP-binding and dynamic positioning of C46SBchN, C153SBchN, C220SBchN, C309SBchN and C217SBchB, respectively.

We, furthermore, expect two molecules of ATP to bind to the wild-type enzyme. Combining the mutagenesis data with kinetic measurements, we propose that at least four molecules of ATP are required to overcome the kinetic barrier to the two-electron reduction of Pchlide. This scenario is in good agreement with the dimeric state of Bchl$_2$, presumably bridged by an intersubunit [4Fe-4S] cluster. The cluster would transfer a single electron to a second intersubunit [4Fe-4S] cluster coordinated by both BchN and BchB within the (BchNB)$_2$ complex. This second cluster then transfers the electron to the substrate, resulting in a reduced Pchlide radical. Consumption of another two ATP molecules by Bchl$_2$ would then allow a second electron to follow the first, resulting in fully reduced Chloride.

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