Evaluating the Suitability of Clays from Abakaliki Area, Southeastern Nigeria for Oil Industrial Application Using Geotechnical and Rheological Properties

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Abstract: The characteristics of clays from Abakaliki area, southeastern Nigeria were characterized to establish its suitability in oil industry as drilling mud when compared with naturally active bentonitic clays from Wyoming and Texas which are used in the industry as drilling mud. The chemical, mineralogical and geotechnical properties were employed in assessing the suitability of Abakaliki clays as drilling mud. Mineralogically, the clays were all characterized as dominantly Illite as well as montmorillonite with low percentages of kaolinite. The chemical composition of the clays indicates low percentages of Na₂O when compared with that of Wyoming bentonite with fairly higher percentages of CaO and K₂O than are required for drilling mud clays. The clays are plastic; with liquid limit (LL) of 58.8 to 72.8, plastic limit (PL) of 25 to 30 and plasticity index (PI) of 26 to 45.8. They classify as inorganic clays of high plasticity (CH) according to Unified Soil Classification System (USCS) and as A-7-6 according to American Association of State High and Transportation Official’s (AASHTO) classification systems. The natural pH values are greater than 8.10, while the viscosity values varied from 5.9 centipoises to 8.0 centipoises for 10 g clay per 350 ml water. Some of these natural properties (Na₂O and CaO contents, LL and PI and viscosity) of these Nigerian clays failed to satisfy the required specifications outlined for clays used as drilling mud when compared with the properties of Wyoming bentonite. This clay will possibly yield low plastic viscosities but with additives such as Sodium Carbonate and Carboxymethyl Cellulose, the rheological properties for Abakaliki clay will remarkably improve.

Keywords: Abakaliki Clay, Drilling Mud, Geotechnical Analysis, Southeastern Nigeria, Wyoming Bentonite

1. Introduction

Clay is one of the earliest mineral substances utilized by man. It played an extremely important part in ancient civilizations, records of which were preserved in brick buildings, in monuments and in pottery, and as inscriptions upon clay tablets. Clay are generally fine-grained materials, with particle size less than 0.002mm with predominately clay minerals. Other minerals associated with clay minerals in clays may include quartz and feldspar (Grim, 1968), as well as detrital materials that were eroded from the earth’s surface. Clay is still an indispensable raw material today. The present uses of clay and clay products are too numerous to list completely. Because of its characteristic plastic property, clay has several industrial uses, which include manufacturing of refractories as well as in drilling mud in the water, oil and gas industries. Several clay occurrences have been investigated and reported in Nigeria (Elueze, 1998). The interest in these studies is hinged on the effective exploitation of the local raw materials for use as drilling mud for sustainable economic growth and creation of employment opportunities. Despite the abundance of clay and its wide industrial uses, it is essential that certain property specifications be met by either the raw or refined clay. The refining of clays usually alters the geotechnical properties of the natural clays, hence, could boost their industrial potentials. Raw materials used for the mud making are usually selected clays with appreciable amount of montmorillonite and are judged by their behavior in water (Nestle, 1944). Their suitability is determined by various criteria, among which are the viscosity, the volumetric yield of given clay and filtration characteristics (Apugo-Nwosu et
The raw material which meets most of these requirements for drilling is the bentonite clay. Bentonite is formed by the weathering of volcanic ash. Wyoming bentonite can swell up to 16 times its original size and absorb up to 10 times its own weight in water. Calcium bentonite, a low or non-swelling variety, is relatively unimportant in Wyoming production. The weathering process, by which the clay minerals are formed from the parent minerals are complex but the main factors are climate, topography, vegetation, and time of exposure (Jackson, 1957). In addition to montmorillonite, bentonite may also contain feldspar, biotite, kaolinite, illite, cristobalite, pyroxene, zircon, and crystalline quartz (Parkes, 1982). By extension, the term bentonite is applied commercially to any plastic, colloidal, and swelling clay regardless of its geological origin (Apugo-Nwosu et al., 2011). Such clays are ordinarily composed largely of minerals of the montmorillonite group. Bentonite is clay that expands to many times its original volume when wetted; it is also characterized by a high cation-exchange capacity, by an ability to form viscous suspensions and thixotropic gels when mixed with large amounts of water, by a high adsorption capacity, and by an ability to form an impervious seal (Rath, 1986).

Drilling activities for the development of oil, gas and water resources usually require enormous amount of drilling mud, which is usually imported into the country. Such importation, apart from draining the nation’s reserves in hard currency, is also inimical to local content policy being currently promoted for the oil industry in the country. Thus, the need for local sourcing of bentonitic clays or close substitutes has become imperative. Fortunately, clay deposits occur abundantly in the sedimentary basins of Nigeria, especially in Abakaliki area (Figure 1).

Research performed in the early 90’s by the Bureau of Mines of the U.S.A., showed that the sum of bentonite deposits in the world was about 1.36 billion tons, and the U.S.A. has more than 50.0% of the total (Apugo-Nwosu et al., 2011). Clays of various kinds and grades abound throughout Nigeria’s sedimentary basins and on the basement (Falode et. al., 2007). Recent investigations by the Nigerian Mining Corporation established the existence of bentonitic clay reserves of over 700 million tonnes in the country, with the largest single deposit at Afuze in Edo State holding 70–80 million tonnes (Apugo-Nwosu et al., 2011).

Although there is a general absence of universally accepted procedure for standardizing industrial specifications of the properties of clays used as drilling mud, the properties of the naturally active Wyoming bentonite which most oil drilling companies use as an effective drilling mud have been employed to evaluate the potential of most clays applied in drilling operations (Rath, 1986). Thus, earlier workers such as Omole et al. (1989) were able to have a pioneering study on the suitability of the black cotton soil clay of northeastern Nigeria as drilling mud. Their work evaluated the nature and geotechnical properties as well as effects of beneficiation on the clay soil. The results of their beneficiation tests showed that the beneficiated clays were still not suitable for direct use as oil drilling mud but could be used for drilling water wells. Later researchers, such as Mihalakis et al. (2004), observed that the rheological properties of the black cotton clays improved with addition of Na₂CO₃, such that the upgraded clay can be used as multipurpose drilling mud.

Other researchers (Ezeribe and Oyedeji, 2005) added lignite to clay mud to improve its rheological and filtration properties. The lignite was activated with sodium hydroxide and added to fresh water mud at different concentrations. The authors concluded that the addition of lignite improved the suitability of the clay with a significant improvement observed when 3% of lignite was introduced. The work of Okogbue and Ene (2008) revealed that some southeastern Nigeria natural clays possess properties that are somewhat similar to those of naturally active bentonitic clays from Wyoming and Texas which are used in the industry as drilling mud. Physical and rheological properties of the clays were however, said to be poor. The characteristics of Abakaliki clays for use as drilling mud have not been evaluated.

This work evaluates, by means of simple but relevant geotechnical tests, the rheological properties of some clay deposits from Abakaliki area, southeastern Nigeria. The aim was to determine whether the clay deposits meet the specifications for use as drilling mud when compared with that of Wyoming bentonite, as well as other commercial bentonite.

### 1.1. Literature Review

### 1.2. Properties Required for Drilling Mud

#### 1.2.1. Chemical Properties

The chemical properties of drilling mud are central to the performance and hole stability. Properties that must be anticipated include the dispersion of formation clays or dissolution of salt formations; the performance of other mud products- for example, polymers is affected by pH and calcium; and corrosion in the well. Sodium montmorillonite particles are laminar in shape carrying surface electrical charge not homogeneously distributed. The faces bear a
permanent negative charge while the edges are conditionally charged depending on the pH of medium (Kelessidis et al., 2005). It is the imbalance in the electrical charge that characterizes the type of interactions of clay particles with other materials.

1.2.2. Rheology and Density

In geology, rheology is defined as the study of how matter deforms the flows, including its elasticity, plasticity and viscosity. Rheology is an extremely important property of drilling fluids, work-over and completion fluids, cements and specialty fluids and pills. Mud rheology is measured on a continual basis while drilling and adjusted with additives or dilution to meet the needs of the operation. A high viscosity fluid is desirable to carry cuttings to surface and suspend weighting agents in the mud. However, if viscosity is too high, friction may impede the circulation of the mud causing excessive pump pressure, decrease the drill rate, and hamper the solids removal equipment. Plastic viscosity is a measure of the internal resistance to fluid flow attributable to the amount, type and size of solids present in a given fluid (Garvey et al., 1988). The value expressed in centipoises, is proportional to the slope of the consistency curve determined by the region of laminar flow for materials obeying Bingham’s Law of plastic flow.

1.3. Geomorphology

Abakaliki which is the study area as shown in Figure 2 is made up of villages like Nkaliki, Onuebonyi, Aghaja Unuhu, Ugbo, Ishieke, Izzi unu and Abakaliki urban. The area is geographically located between latitude 6°15’N and 6°20’N and longitude 8°05’E and 8°10’E covering a total area of about 81km². The areas were accessed through Enugu, Ogoja, and Afikpo through a network of tarred roads which include the Abakaliki-Enugu Express road, Abakaliki-Ogoja Express road and Abakaliki-Afikpo Express road. During the fieldwork, vehicles, motorcycles and footpaths were employed to aid accessibility and movement through the outcrop locations and sample collections. The vegetation of Abakaliki and its environs is luxuriant vegetation of tropical rainforest.

The Abakaliki area lies within one of the nine vegetation zones in Nigeria as shown in Figure 3. Regionally the vegetation type is derived savanna. According to Offodile (1992), topography, drainage and rainfall controlled the vegetation. The vegetation type of the Abakaliki area is parkland; this is characterized by stunted trees and pockets of derelict woodland and secondary forest consisting of few shrubs with dispersed large trees and climbers. Its vegetation is densely populated with grasses and trees of different sizes in the area. The area is marked by undulated range of shale outcrops and the shales are either greyish or reddish brown in color depending on its content and degree of weathering. The area had 400ft as its highest contour and 100ft as its lowest contour above sea level. The climate of the study area is of humid tropical climatic region. It experiences one rainy season and one dry season (eight months of rainfall and four months of dryness). Harmattan is felt between December & January. The mean annual temperature stands at 28°C. Humidity in this area is about 50-60% Pa. The mapped area has a mean annual rainfall of 2500mm. Rainfall is highest in September and lowest in January. The drainage system of the study area is dendritic in pattern, as a function of lithologic control. The study area is mainly drained by Iyiokwu River, Iyiudene River and Ebonyi River with few minor drainage flows. All these, both the major and minor drainage system flow eastward to join the Cross River Somewhere outside the study area. There are two types of settlements observed in the study area. They are dispersed and nucleated settlement. The dispersed settlements were found in Idembia Enyichiri, Ekaun Inyimagu, and Agharaugo Umuoghara areas. In these areas, dwelling places are scattered and isolated from each other while the nucleated settlements were observed in Abakaliki urban and Ishieke.
2. Geology

Abakaliki southeastern Nigeria as shown in Figure 4 is underlain by the southern Benue Trough (Reyment, 1965). The origin of the Benue Trough is closely associated with the breakup of western Gondwanaland during the separation of the African and South American Plates, and opening of the south Atlantic Ocean, in the early Cretaceous (Wright, 1976). The Benue Trough has been described as an ‘intracratic rift system’ or an ‘intercontinental Cretaceous Basin’ stretching in a NE–SW direction (Benkhelil et al., 1989). The southern section of the trough, which underlies most parts of southeastern Nigeria, has stratigraphic record of deposits represented by sediments of three main marine depositional cycles, Albian-Cenomanian, Turonian-Santonian, and Campano- Maastrichtian (Reyment, 1965). The first marine transgression in the trough is generally believed to have started in the mid-Albian period with the deposition of the Asu River Group made up of predominantly shales and localized development of sandstone, siltstone and limestone facies (Hoque and Nwajide, 1988). The study area is covered by sediments of Asu River Group. The Abakaliki Shale Formation, which has an average thickness of about 500m, is dominantly shale, dark grey in colour (weathers to brownish material in the some part of the Formation), blocky, and non-micaceous in most locations. Other authors agreed that the shales may have been affected by low-grade metamorphism which occurred during the Santonian. It is calcareous (calcite-cemented) and gives off effervescence on contact with dilute hydrochloric acid (Okogbue and Aghamelu, 2010). Table 1 gives the distributions of the Asu River Group and other major geological formations of the southern Benue Trough. The regressive phase of the first marine transgression led to the deposition of the Cenomanian sediments. The beds of this age are located in the Southeastern part of Nigeria particularly around Calabar. These beds have been assigned as the Odukpani Formation (Reyment, 1965). The Formation according to Reyment (1965) consists of arkosic sandstones, limestones and alternating limestones and shales which became gradually more predominantly shaly in its uppermost parts.

![Figure 4. Geological map of Nigeria showing Abakaliki the study area (Modified after Benkhelil et al., 1989).](image)

### Table 1. Regional stratigraphic sequence of southeastern part of Nigeria. (Modified from Reyment, 1965 and Murat, 1972).

| AGE      | FORMATION | SEDIMENTARY CYCLE         |
|----------|-----------|---------------------------|
| Pliocene | Benin Formation | Niger-Delta basin                  (Third sedimentary cycle) |
| Pleistocene | Ogwash-Asaba Formation | Anambra-Afikpo Basin (Second sedimentary cycle). |
| Eocene | Ameki Formation | |
| Paleocene | Imo Shale | |
| Maastrichtian | Nsukka Formation | Abakaliki-Benue Basin (first sedimentary cycle). |
| Campanian | Ajali Sandstone | |
| Santonian Coniacian | Mamu Formation | |
| Turonian | Awgu Shale | |
| Cenomanian | Ezeaku Formation | |
| Albian | Asu-River Group (Abakaliki clay) | |

3. Materials and Methods

3.1. Sampling

Ten (10) fresh samples of clays were taken from Abakaliki area in southeastern Nigeria. The samples are herein designated as L1 to L10. Care was taken when sampling to ensure that fresh samples were taken and that such samples were representative of the materials to be assessed. The sampling depth ranged from 1 m to 2.5 m. Tests on the clay samples were carried out at the material laboratories of the National Steel Raw Material Exploration Agency, Kaduna in Nigeria.

3.2. Laboratory Tests

Relevant laboratory tests were carried out to determine the properties of the sampled Abakaliki clays. The pulverized clay sample was prepared by creating a highly polished surface in the cavity of the sample holder of the Pw 1800 diffractometer with copper tube anode. This was introduced to the XRD equipment which scanned the sample
continuously for clay bulk analysis. A generator tension of 40kv and current of 55mA were used. From the XRD traces, the clay minerals in the samples were determined by their diagnostic peaks according to diffraction intensity published by ASTM (1989), in powder diffraction files. An IVT-20 computer-automated X-ray fluorescence (XRF) spectrometer was used for bulk chemical analysis of the clay samples. The clay samples were prepared in accordance with outline given in Kenigan (1971). Cation Exchangeable Capacity of metallic bases in meq/100 g of dry sample were measured by leaching 50 g of the sample with neutral normal ammonium acetate and the resulting aliquots were analyzed characteristically for Na, K, Ca and Mg cations, following method outlined in ASTM 2354 (1989). The fraction that passed the 75 µm BS (No. 200) sieve was oven-dried, weighed and later dispersed with a solution of sodium hexametaphosphate and sodium carbonate. It was then analyzed using the hydrometer method as specified in BS 1377 (1990). The bulk density and specific gravity were determined using 100 g of oven-dried samples, passing 75 µm BS (No. 200), according to procedures outlined in ASTM (1989). The viscosity and pH were both determined from clay-water suspension with 20 parts water using methods described in L ASTM C97 (1989). The Atterberg limits of the clay samples were determined in accordance with BS 1377 (1990). Free swell test was conducted using methods described by Holtz and Gibbs (1956).

4. Results

4.1. Mineralogical, Chemical Analysis and Cation Exchangeable Capacity

The diffractograms of the clay samples are as presented in Figure 5. The results of the mineralogical composition of Abakaliki clays are summarized in Table 2 and compared with that of Wyoming clay, as reported by Falode et al. (2007).

![Diffractograms of the clay samples and Wyoming bentonite.](image)

| Clay Mineral   | Wyoming bentonite* | Abakaliki clay |
|----------------|--------------------|----------------|
| Montmorillonite| 35.85              | 20-30          |
| Kaolinite      | 5.01               | 15-25          |
| Quartz         | 23.98              | 28-30          |
| Calcite        | 8.32               | 13-24          |
| Biotite        | 3.23               |                |
| Feldspar       | 23.98              | 28-32          |
| Illite         | 0                  | 30-38          |

The results indicate that the clays are composed predominantly of Illite and mixed layer clays, with presence of Montmorillonite which ranges from 20-30 and kaolinite which ranges from 15-25. Montmorillonite produces “sensitive” clay even where it is present in minute (5% to 10 %) amount (Okagbue, 1989). The presence of the montmorilloite in Abakaliki is significantly low when compared with Wyoming bentonite. The presence of quartz, which acts as an abrasive according to Apugo-Nwosu et al. (2011), could lead to poor physio-chemical performance of the clay. It has been observed (Sowers and Sowers, 1970) that Na-montmorillonite is quite more expansive than the Ca-montmorillonite type. The presence of kaolinite in the Abakaliki clay could lead to poor rheological properties since such type of clay according to Apugo-Nwosu et al. (2011) has low swelling capabilities.
4.2. Chemical Composition

The results of chemical analysis of the clay samples are presented in Table 3. The chemical composition of the Abakaliki clay sample indicated very high alumina (19.90-25.08%) and silica (50.10-58.96%) when compared with the value of Wyoming bentonite that has silica of 45% and alumina of 17% (Table 2) as well as standard commercial bentonite which has 48.80% silica and 15.54% alumina. The alkaline earth mineral (CaO) and the alkali mineral (K₂O) of Abakaliki clay are 1.00-5.42% and 0.52-1.40% respectively and that of Wyoming bentonite are alkaline earth mineral 1.17% and alkali mineral (0.05%) respectively. The table indicates low content of Na₂O (0.68-1.98%) when compared with that of Wyoming clay which has 2.7 of Na₂O. The fair consistency in the concentrations of oxides may generally indicate similarities in the mineralogy of the studied clays. The relatively high amounts of CaO noted in the clays may possibly be attributed to the occurrence of carbonates such as calcite in the sediments (Reyment, 1965). In all, the enrichment in CaO, as well as depletion in Na₂O, might suggest that the predominant clay mineral is the Ca-rich type. The presence of alkalis and magnesia in the samples suggests presence of clay mineral which may not likely be montmorillonite. Abakaliki clay showed low value of Fe₂O₃, this indicates low laterite concentrations. The very high alumina and silica contents when compared with refractory standards (25-45% alumina and 55-75% silica) as noted by Falode et al. (2007), coupled with the high levels of iron and organic matter would make the Abakaliki clay sample very suitable as a refractory material.

The MgO value for Wyoming clay and standard commercial bentonite (Table 3) is very close to the value recorded for Abakaliki clay. This compound is normally used to enhance gel strength of mud samples (Falode et al., 2007). The results of the analysis of the samples for the CEC are presented in Table 4. Ca²⁺ is the main cation in all the studied clays while Na⁺ is a minor cation. The CEC recorded for Wyoming bentonite indicates that the values range from 60-65 Na⁺ to 15-20 Ca²⁺. The typical CEC range of pure smectite is from 80 meq/100 g to 150 meq/100 g (Grim, 1968). The presence of Na and Ca ions, among other factors, influences the swell potentials of active clays. While Ca²⁺ limits the swelling of montmorillonite, Na⁺ enhances it. Tourtelot (1974) had observed that Ca²⁺ as the main exchangeable cation has a strong depressing effect on the swelling capacity of montmorillonite. Ca-montmorillonites can swell only up to 100% while Na-montmorillonite in bentonite may swell up to 2000%. Natural clays rich in montmorillonite generally contain a mixture of exchangeable cations, including Mg²⁺, K⁺ and Fe³⁺, although Na⁺ and Ca²⁺ predominate but in varied proportions. Montmorillonite with these mixed proportions of exchangeable cations would be expected to swell in amounts intermediate between those of Ca-montmorillonite and Na-montmorillonite end members (Tourtelot, 1974).

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### Table 3. Chemical Composition of Abakaliki clay.

| Chemical Oxides | Abakaliki clay* | Standardcommercial bentonite | Wyoming Bentonite |
|-----------------|-----------------|------------------------------|-------------------|
| SiO₂            | 50.10-58.96     | 48.80                        | 45.0              |
| Al₂O₃           | 19.90-25.08     | 15.54                        | 17.0              |
| Fe₂O₃           | 3.80-4.67       | 6.44                         | 11.1              |
| TiO₂            | 1.10-2.10       | 0.49                         | 1.68              |
| CaO             | 1.00-5.42       | 5.22                         | 1.77              |
| MgO             | 0.32-2.02       | 3.50                         | 0.33              |
| K₂O             | 0.52-1.40       | 0.75                         | 0.05              |
| Na₂O            | 0.68-1.98       | 2.19                         | 2.70              |
| MnO             | 2.00-3.34       | 0.07                         | 0.15              |

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### Table 4. Results of Cation exchangeable capacity and hydrometer test.

| Cation Exchangeable Capacity meq/100 g | L1  | L2  | L3  | L4  | L5  | L6  | L7  | L8  | L9  | L10 | Ave.  |
|--------------------------------------|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-------|
| Ca²⁺                                 | 33.2| 36.3| 38.6| 32.3| 30.1| 32.9| 28.8| 34.8| 35  | 28.98| 33.1  |
| K⁺                                   | 1.49| 2.36| 2.07| 2.69| 3.66| 2.12| 2.31| 2.07| 2.67| 2.19 | 2.4   |
| Na⁺                                  | 12.3| 16.6| 12.8| 11.1| 13.5| 12.4| 16.5| 12.3| 14.8| 12.79| 13.5  |
| Mg²⁺                                 | 4.3 | 5.7 | 5.3 | 4.9 | 5.6 | 4.9 | 6.7 | 5.3 | 5.4 | 6.7  | 5.5   |
| Results of Hydrometer Analysis %     |     |     |     |     |     |     |     |     |     |     |       |
| Sand                                 | 18  | 14  | 10  | 8   | 10  | 20  | 29  | 20  | 14  | 11   | 15.4  |
| Silt                                 | 48  | 46  | 44  | 52  | 48  | 42  | 38  | 48  | 48  | 51   | 46.5  |
| Clay                                 | 34  | 38  | 46  | 40  | 42  | 38  | 41  | 42  | 38  | 38   | 39.7  |
| Viscosity in centipose               | 7   | 6   | 8   | 6.5 | 7.8 | 6.9 | 8.9 | 8   | 8.3 | 6.7  | 7.4   |
| Free Swell%                          | 60  | 63  | 59  | 55  | 62  | 56  | 60  | 50  | 58  | 58   | 58.1  |
| Ph                                   | 8.1 | 9   | 8.25| 9.1 | 8.18| 9.3 | 7.8 | 7.95| 8.12| 7.9  | 8.37  |
4.3. Geotechnical Properties

The results of the Atterberg limits tests on the natural are presented in Table 5. These results indicate that the clays in their natural states have relatively high plasticity. When compared with Wyoming bentonite which has LL 600-700 (Skempton, 1953), the LL values (58.8-72.8) of Abakaliki clay, however, are significantly low. Atterberg plasticity index (API) of 26-45.8 shows that the sample is plastic over a wide range of moisture content. These also corroborate the alkaline earth mineral of 1.00-5.42% and the alkali mineral of 0.52-1.40% of Abakaliki clay.

The high plasticity index using plasticity chart (Skempton, 1953) reveals the clay’s potential for great volume change characteristics. The silt content makes the sample material unsuitable for wide ceramic applications. The results of the gradation analysis and the calculated Skempton activity values of the clay samples are also presented in Table 5. The results indicate that the samples are characterized by marginally higher proportion of silt-sized particles with average value of 46.5% than the clay-sized particles with average value of 39.7%. The clay fractions, despite being lower than the silt fraction, would be considered significant and are capable of influencing the geotechnical behaviour of the clays, especially if they were made of the highly expansive mineral type. Okagbue (1989) had noted that the colloidal (<0.001 mm) content of a soil is an important influencing factor in the absorption of water and in volume changes in the soil depending on the mineral types constituting the colloids.

Activity (A) according to Skempton (1953) of the soil is the ratio of the plasticity index and the percentage of the clay fraction. Activity is a measure of the water-holding capacity of the clayey soils. The Activity values of the tested samples which range from 0.63 to 1.21 may suggest that the clays are possibly Ca-rich, rather than Na-rich. Soda rich active clays, according to Skempton (1953) and Tourtelot (1974), have Activity values greater than 5. The average activity value for the studied clay is 0.92 (see Table 6) indicating normal soil type. Active clays have both high water retaining and high cation exchangeable capacity (Skempton, 1953).

Table 5. Results of Atterberg limit test, specific gravity, bulk density and calculated Activity value.

| Parameters       | L1    | L2    | L3    | L4    | L5    | L6    | L7    | L8    | L9    | L10   | Ave.  |
|------------------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
| Specific gravity | 2.54  | 2.56  | 2.52  | 2.42  | 2.64  | 2.6   | 2.2   | 2.45  | 2.6   | 2.54  | 2.34  | 2.48  |
| Bulk density     | 1.55  | 1.68  | 1.67  | 1.52  | 1.52  | 1.45  | 1.7   | 1.57  | 1.59  | 1.45  | 1.53  | 1.57  |
| Plastic Limit    | 24.8  | 24.6  | 30.1  | 25    | 27    | 34    | 27    | 30    | 32    | 28    | 28.25 |
| Liquid limit     | 66    | 58.8  | 70    | 62.4  | 72.8  | 65    | 62    | 56    | 60    | 63    | 63.6  |
| Plasticity index | 41.2  | 44.2  | 39.9  | 37.4  | 45.8  | 31    | 35    | 26    | 32    | 35    | 36.75 |
| Activity         | 1.21  | 1.16  | 0.86  | 0.86  | 0.86  | 0.81  | 0.85  | 0.63  | 0.84  | 0.92  | 0.92  |

*Average Activity value

Table 6. Classification of soils based on Activity.

| Activity | soil type       | Abakaliki clay |
|----------|-----------------|----------------|
| A<0.75   | Inactive        |                |
| A= 0.75 to 1.25 | Normal        | 0.92*          |
| A> 1.25  | Active          |                |

The swelling potential values as indicated from free swell as shown also Table 4 also suggest that the clays are not the very active types. Using LL as one of the swell indicators (Chen, 1988), the free swell test results also indicate that the tested natural clay samples have low swelling potentials when compared with bentonites from Texas with mean LL value of 1110% and free swell range of 397 and 401 (Rath, 1989). As shown in the tables, there is fair consistency in the results of bulk density of all the tested clays, suggesting similarity in the material type; a situation already reflected in the results of mineralogical and chemical analyses. The result suggests that the clays generally have low Gc values (average of 2.48) when compared with that of Texas bentonitic clay which has specific gravity range of 2.50 to 2.80g/cm³. Commercial Bentonite sample of China origin according to (Abdullahi, et al., 2011) has specific gravity of 2.87. Kearey, (2001) noted that a crystal of montmorillonite would have Gc of about 2.74. Since tests have suggested that the tested clays are illite predominated, it may mean that the other fraction that is non-montmorillonite is, most probably the mixed layer fraction, of the low Gc clay mineral type. Attapulgite (or palygorskite) is to be suspected because of its close association with Ca-rich montmorillonitic clays (Kearey, 2001) and its low Gc, (mean Gc value of 2.48); Lambe and Whitman (1979). The clay needs increase in its density since according to Hall (1993), clays used as drilling mud must be “weighted” by addition of high density solids such as barite or haematite in order to control over pressure.

Figure 6. Location of clay minerals on the Casagrande plasticity chart using the Abakaliki clay (Modified from Skempton, 1953).
4.4. Rheological Properties

The results of the pH measurements of the clay slurries indicate that they are all alkaline. The viscosity of the natural clay slurries, also shown in Table 4, range from 6.0 to 8.9 centipoises. These values are comparatively low when compared with those of drilling mud. Rath (1986) observed that natural bentonitic clays from Wyoming recorded viscosity that ranged between 8.0 centipoises and 25.0 centipoises. Bentonites with higher clay percentages (lower non-clay mineral content) as Wyoming should show high rheological properties. Clay such as montmorillonite that has a high cation exchange capacity swells greatly and forms viscous suspensions at low concentrations of clay, particularly when sodium is in the exchange position. Considering the relative higher swelling volumes of Wyoming bentonite indicated by higher montmorillonite content, it should make better suspensions with water and therefore good rheological properties can be obtained whereas the Abakaliki clay would require some beneficiation to improve the swelling properties. This clay will possibly yield low plastic viscosities but with additives such as Sodium Carbonate and Carboxymethyl Cellulose, the rheological properties for Abakaliki clay will remarkably improve.

### Table 7. Comparison of properties of the Abakaliki clays with those of natural sodium bentonites and activated varieties (Modified after Rath, D.L., 1986 and Okogbue, et al., 2011).

| Parameter                      | Natural sodium bentonite | Activated Varieties | Abakaliki clay (average) |
|--------------------------------|--------------------------|---------------------|--------------------------|
|                                | Wyoming                  | Texas               | Manitoba                 | India                    |
| SiO₂                           | 58-64.00                 | 63.5-64.90          | 63.70                    | 66.10                    |
|                                |                          |                     |                          | 50.10-58.96*             |
| Al₂O₃                          | 18-21.00                 | 9.26-9.32           | 19.90                    | 13.10                    |
|                                |                          |                     |                          | 19.90-25.08*             |
| Fe₂O₃                          | 2.5-2.80                 | 2.52-2.57           | 1.40                     | 5.45                     |
|                                |                          |                     |                          | 3.80-4.67*               |
| MgO                            | 2.5-3.20                 | 1.79-1.80           | 4.60                     | 1.57                     |
|                                |                          |                     |                          | 1.10-2.10*               |
| CaO                            | 0.1-1.00                 | 0.83-0.88           | 0.16                     | 0.58                     |
|                                |                          |                     |                          | 1.00-5.42*               |
| Na₂O                           | >2.70                    | 4.03-4.06           | 0.77                     | 0.09                     |
|                                |                          |                     |                          | 0.32-2.02*               |
| K₂O                            | 0.2-0.40                 | 1.20-1.28           | 0.26                     | 1.40                     |
|                                |                          |                     |                          | 0.52-1.40*               |
| TiO₂                           | 0.1-0.20                 | +                   | 0.12                     | 0.17                     |
|                                |                          |                     |                          | 0.68-1.98*               |
| Density (g/cm³)                | +                        | 2.5-2.80            | +                        | +                       |
|                                |                          |                     |                          | 1.57                     |
| Viscosity                      | 8.25                     | +                   | 10.00                    | 12.00                    |
|                                |                          |                     |                          | 8.37                     |
| Specific gravity               | 2.70                     | 2.80                | 2.60                     | 2.63                     |
|                                |                          |                     |                          | 2.48                     |
| pH                             | 8.5-10.00                | 8.10                | 5.60                     | 7.80                     |
|                                |                          |                     |                          | 7.4                      |
| Free swell %                   | 950                      | 1110                | 700                      | 720                      |
|                                |                          |                     |                          | 58.1                     |
| Atterberg Limits               |                          |                     |                          |                          |
| LL                             | 600-700                  | 397-401             | +                        | +                       |
|                                |                          |                     |                          | 63.6                     |
| PL                             | +                        | 38                  | +                        | +                       |
|                                |                          |                     |                          | 28.25                    |
| PI                             | +                        | 358-362             | +                        | +                       |
|                                |                          |                     |                          | 36.75                    |
| CEC (meq/100 g)                |                          |                     |                          |                          |
| Na⁺                            | 60-65                    | -                   | 63                       | 53                       |
|                                |                          |                     |                          | 33.1                     |
| Ca²⁺                           | 15-20                    | +                   | 25                       | 14                       |
|                                |                          |                     |                          | 2.4                      |
| Mg²⁺                           | 5-10                     | +                   | 15                       | 4                        |
|                                |                          |                     |                          | 13.5                     |
| K⁺                             | 1-5                      | +                   | 8                        | 3                        |
|                                |                          |                     |                          | 5.5                      |

Note: *Range value, + not available

4.5. Performance of the Clays as Drilling Mud

A good drilling mud should have excellent plasticity, lubricity that protects the drilling bit, and viscosity that is required for the transportation of drill cuttings. A comparison of the properties of the Abakaliki clay and the properties of Wyoming bentonite as shown in Table 7 indicates that the recorded viscosity is low against the standard clays that have their viscosity values ranging between 8 centipoises and 25 centipoises. With respect to the chemistry, the concentrations of most of the oxides generally compare well with those of the standard clays. Few oxides, such as Al₂O₃, MgO, TiO₂, however are marginally higher. Texas clay recorded bulk density that ranges from 2.50 g/cm³ to 2.80 g/cm³ the Abakaliki clay has their bulk densities all below 1.73 g/cm³. A significant inadequacy is also observed in the Atterberg limits, swelling potentials and other properties linked to response of clay minerals to moisture conditions. For example, the free swell values of the Abakaliki clays are generally below 63 %, whereas the standard clays all have free swell above 700 %. Similar trend is also noticed for PI. While the natural studied clays all have API below 46 the standard clays record PI all above 358. Exchangeable Ca²⁺ is also quite higher for all the samples studied in comparison with that of the standard clays.

5. Conclusions

This study evaluated the physical and chemical properties of some Abakaliki clays from Abakaliki Formation, southeastern Nigeria for possible use of the clays as drilling mud. X-ray Diffraction test on the studied clays from Asu River Group showed that the dominant clay minerals are illite, kaolinite and mixed layer types. The chemical compositions indicate that the values of Na₂O are lower than...
would be expected for clays that are used as drilling mud. Ca$^{2+}$ and Na$^{+}$ are the main exchangeable cations with calcium ion as the dominant cation. The clays exhibited presence of high concentration of exchangeable Ca$^{2+}$. The clays in their natural state possess relatively high liquid and plastic limits and classifying as clays of high plasticity (CH), according to Unified Soil Classification System. The weak swelling properties are thought to buttress the presence of low expansive montmorillonite, most probably the calcium variety. A comparison of the chemical composition of the studied clays with that of naturally active Bentonite from Wyoming and Texas used in drilling mud showed that the plasticity and swelling properties of the Abakaliki clays are very low. The viscosity values of the clays are also low. This clay will possibly yield low plastic viscosities but with additives such as Sodium Carbonate and Carboxymethyl Cellulose, the rheological properties for Abakaliki clays will remarkably improve. However, the G$'$, and pH values of the clays are close to those of Wyoming and Texas bentonites.

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