Purification of Human Erythropoietin*

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Human erythropoietin, derived from urine of patients with aplastic anemia, has been purified to apparent homogeneity. The seven-step procedure, which included ion exchange chromatography, ethanol precipitation, gel filtration, and adsorption chromatography, yielded a preparation with a potency of 70,400 units/mg of protein in 21% yield. This represents a purification factor of 930. The purified hormone has a single electrophoretic component in polyacrylamide gels at pH 9, in the presence of sodium dodecyl sulfate at pH 7, and in the presence of Triton X-100 at pH 6. Two fractions of the same potency and molecular size, by sodium dodecyl sulfate gel electrophoresis, but differing slightly in mobility at pH 9, were obtained at the last step of fractionation. The nature of the difference between these two components is not yet understood.

Erythropoietin is an acidic glycoprotein that is present at a very low concentration in plasma under normal conditions. Under anemic or anoxic stress, it is found in relatively large amount in the plasma and is also excreted in the urine. Erythropoietin is the substance that is responsible, in large part, for the regulation of normal red blood cell differentiation. Because of this function, and because it may have a role in replacement therapy of some kinds of anemia, it is important to have pure erythropoietin in an amount sufficient for chemical characterization. Reports on the purification of human (1) and sheep (2) erythropoietin have been published. In the former, the evidence for homogeneity was not convincing, and in the latter, the total amount was too low for adequate characterization. We report in this paper on the preparation of milligram quantities of human urinary erythropoietin in a state of apparent homogeneity.

EXPERIMENTAL PROCEDURES

Bioassay—The fasted rat method of bioassay (3), in which the incorporation of labeled iron into circulating red cells is measured, was used routinely to quantitate the amount of erythropoietin activity. Samples for assay were dissolved in 0.1% bovine serum albumin in 0.15 M NaCl, 0.01 M CaCl₂. Over the 18-month period covered by this work, the In dose-ln response curve obtained when 1, 1.5, 2, and 3 units of erythropoietin/unit rat were used had the following characteristics: slope, 1.13 ± 0.34; intercept, 0.75 ± 0.39; correlation coefficient, 0.96 ± 0.10. The assay values found for the two final hydroxyapatite fractions were confirmed by the polycythemic mouse method (3) which agreed closely with the other two assay methods. We are indebted to Dr. Walter Fried of the Michael Reese Hospital for doing the mouse assays. For this purpose and for the assay of the activity recovered from polyacrylamide gels, biological activity was measured by the marrow cell culture method (4). This procedure, in which both the total uptake of radioiron and its incorporation into hemoglobin are used as quantitative indicators of erythropoietin activity, is about 1000 times more sensitive than the fasted rat method, but does not distinguish between native erythropoietin and the asialo form, which is inactive in vivo.

Materials—Sodium dodecyl sulfate and DEAE-agarose were bought from Bio-Rad Laboratories, Richmond, Calif., and was hydroxylapatite (Bio-Gel HT, Control 12746); we found no significant difference between several different lots which we used. Sulfoethyl Sephadex (Lot 7953) and Sephadex G-100 (Lot 5011) were bought from Pharmacia Inc., Piscataway, N. J. Materials for gel electrophoresis (acrylamide, N,N,N',N'-tetramethylethylenediamine and N,N',N'N'-methylenediaacrylamide) and Triton X-100, scintillation grade, were bought from Eastman Kodak Co., Rochester, N. Y. Labeled iodide was obtained from Amersham-Searle Corp., Arlington Heights, Ill. Other reagents used were of the best quality commercially available.

Ultrafilters were bought from Amicon Corp., Lexington, Mass. PBS is used to designate a solution consisting of 0.15 M NaCl, 0.01 M sodium phosphate buffer, pH 7.0.

Iodination—Labeling with ¹²⁵I (5, 6) was done as follows. To 20 µl of an erythropoietin solution containing 20 µg of protein, 2 µl of 0.5 M Na₂S₂O₃, 2 µl of 1 M NaCl, 0.4 µl of 0.1 M Na₂HPO₄, and 20 µl of dimethylsulfoxide were added. One microliter of Na¹²⁵I (100 µCi; equivalent to 7.14 ng of iodide or 57 pg atoms) was then added, followed by 1 µl of freshly prepared chloramine-T (10 mg/ml in water). The mixture was allowed to stand at 24°C for 10 min, after which 10 µl of Na₂SO₄, (25 mg/ml in water) were added. The solution was mixed and allowed to stand for 1 min; then 200 µl of KI (10 mg/ml in 0.05 M phosphate, pH 7.4) were added and mixed for 1 min at 24°C, followed by addition of 50 µl of 7% (w/w) bovine serum albumin. The mixture was put on a Sephadex G-10 column (25 x 0.9 cm diameter), which had been equilibrated with PBS, being washed over to the column with two 200-µl washes of KI solution (10 mg/ml). The erythropoietin was separated from unreacted iodide by elution with PBS and collection of 0.3-ml fractions. The major peak material of large molecular weight label (Tubes 18 to 28) was pooled and dialyzed. The final volume of 4.1 ml contained 5.8 X 10⁶ cpm of ¹²⁵I (2.9 X 10⁶ cpm/µg of protein, equivalent to 0.1 g atom of iodine/mole of protein).

Because our previous experience showed that sheep erythropoietin was completely inactivated upon iodination using chloramine-T, we used the method of Stagg et al. (5), in which the presence of

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1 One unit of erythropoietin is defined as the biological activity present in one-tenth of the contents of an ampule of the International Reference Preparation distributed by the World Health Organization. In the routine assay, we used, as a working standard, a preparation of sheep erythropoietin that had been standardized against the International Reference Preparation.
dimethylsulfoxide acts to protect methionine residues from oxidation. At the level of 0.1 atom of iodine/molecule, we found no loss of biological activity when the assay was done within 1 day of iodination. One week later, however, there was appreciable loss (36%), indicating that the procedure had labilized the hormone. At a higher degree of iodination (4 atoms/molecule), 75% of the biological activity was lost within 1 day.

Electrophoresis — Polyacrylamide gel electrophoresis was done by the micromethod which we had used earlier (2), with gels that were 5 × 0.2 cm in diameter. The conditions are given below for each experiment. Gels were fixed in 25% isopropyl alcohol, 10% acetic acid overnight, stained in 0.25% Coomassie blue in 10% acetic acid for 1 h, and destained in 10% acetic acid.

Source of Erythropoietin — Urine, 2550 liters, was collected from two groups of patients with aplastic anemia of unknown origin, in several hospitals in Kumamoto City, Japan. These groups included some patients with moderately severe, chronic anemia for whom the urine titer was about 1 unit/ml, and others with severe, chronic anemia for whom the urine titer was 2 to 6 units/ml. The urine was collected in 11 pools and filtered under suction; 2.5-liter batches were collected as before and washed twice with 10 ml of 65% ethanol, 0.5 M LiCl, and the supernatants were pooled. The washed precipitate was dissolved in 20 ml of PBS (65% precipitate).

To the pooled supernatants, 96 ml of ethanol were added slowly, and stirring was continued for 30 min, after which the precipitate was allowed to settle for 14 h at 4°C. We have found that this long period in 75% alcohol is required for optimal further fractionation. The precipitate was washed twice with 10 ml of 75% ethanol, 0.5 M LiCl, the supernatants were pooled, and the precipitate was dissolved in 20 ml of PBS (75% precipitate).

The combined supernatant was brought to 90% ethanol by addition of 540 ml of absolute alcohol, stirred for 30 min, and stored at −20°C for 48 h before the precipitate was collected, dissolved in 50 ml of cold water, and immediately frozen (90% precipitate). The results of one representative ethanol fractionation procedure are given in Table I.

For the 16 experiments, the range of yields in the 90% ethanol precipitate was 28 to 100%. The range of potency was 133 to 650 u/A. Since a substantial fraction of the activity was found in the earlier alcohol precipitates, we established conditions for recovery of much of that activity at a potency similar to that of the 90% ethanol precipitate. For example, three groups of pooled fractions (50%, 65%, and 75% alcohol precipitates), with a total volume of 210 ml, were stirred at 4°C while 120.37 g of guanidine hydrochloride were added. To the clear solution, 52.5 ml of 10 M LiCl were added with continued stirring. The slightly cloudy solution was stirred for 30 min more, and 790 ml of absolute ethanol was added slowly. After 30 min of stirring, the suspension was kept at 4°C for 25 h, and the precipitate was collected by centrifugation at 21,000 × g for 15 min at −15°C, to yield Supernatant A and a pellet. The pellet was suspended in 50 ml of PBS and stirred for 30 min while 28.66 g of guanidine hydrochloride were added. During the next 20 min, 12.5 ml of 10 M LiCl were added, followed by 187.5 ml of absolute ethanol. Stirring was continued for 30 min, after which the suspension was allowed to settle at 4°C for about 14 h. The precipitate was removed at 21,000 × g for 15 min, yielding another 75% ethanol supernatant. The re-extraction was repeated; the two supernatant fractions (Supernatant B) were pooled, but this pool was kept separate from the original Supernatant A. The 75% alcohol precipitate was suspended in 200 ml of PBS and mixed thoroughly.

To Supernatant A, 1062 ml of absolute ethanol were added

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**Table I**

| Ethanol Fractionation | Fraction | Activity u/A | % Yield | Purification |
|-----------------------|----------|--------------|---------|--------------|
|                       | Original | 2,150         | 111,900 | 52           | 100          |
| 50% precipitate       | 570      | 8,900        | 16      | 8            |
| 65% precipitate       | 306      | 15,100       | 53      | 14           | 1.02         |
| 75% precipitate       | 131      | 15,800       | 129     | 15           | 2.5          |
| 90% precipitate       | 173      | 75,800       | 443     | 69           | 8.5          |

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2 Potency, or specific activity, is expressed as units of biological activity (u) per absorbance unit (A), measured at 268 nm in 1-cm cuvettes.

3 This figure is slightly different (1.2% higher) from that indicated as the amount obtained from the DEAE-cellulose step. This kind of difference is caused by the uncertainty in the bioassay and will also be seen at subsequent steps.

4 LiCl was used in the alcohol precipitation procedure in order to increase the solubility of proteins in ethanol (9). Precipitation in the absence of salt resulted in a low potency fraction.
slowly; the suspension was stirred for an additional 30 min and stored at -20°C for 40 h. The precipitate was then collected at 21,000 × g for 15 min at -15°C, washed twice with 40 ml of 90% ethanol, 0.2 M LiCl, and suspended in 30 ml of PBS. After stirring for 30 min, the suspension was centrifuged at 16,000 × g for 10 min at 4°C; the small amount of precipitate was washed with 20 ml of PBS, and the supernatant solutions were pooled (90% ethanol precipitate). The pellet fraction was suspended in 20 ml of PBS. Supernatant B was treated in an analogous manner, i.e., the 90% alcohol precipitate was collected and both Supernatants A and B were assayed. The fraction precipitated at 90% alcohol from Supernatant A contained 246,840 units (54% of the input activity) with a potency of 560 u/A. The fraction precipitate from Supernatant B contained 69,300 units (15% yield) with a potency of 565 u/A. When all of the available ethanol precipitates were re-extracted, we recovered 1,515,200 units with a mean potency of 570 u/A (range, 220 to 680). This material was pooled with the original 90% alcohol precipitates for further fractionation, making a total of 4,750,740 units at a mean potency of 633 u/A.

**DEAE-Agarose Fractionation** — the solution, in water, of a 90% ethanol precipitate was concentrated to about 5 ml on an Amicon UM-10 ultrafilter, then brought to 25 ml with 0.01 M Tris, pH 7.0, and a 50-μl aliquot was removed. The DEAE-agarose, 100 to 200 mesh, was degassed under reduced pressure, suspended in 0.01 M Tris, pH 7.0, and packed into a column 9.2 × 2.5 cm in diameter (bed volume, 45 ml). The gel was washed with 1.5 liters of 0.01 M Tris, pH 6.9; the ratio of absorbance units added to bed volume (ml) was 6.65. The sample was added to the column over a period of 40 min, and 150-drop fractions were collected. The column was washed with 211 ml of 0.01 M Tris, pH 7.0, and then eluted with the following buffers: 366 ml of 0.01 M Tris, pH 7.0; 17 mM CaCl₂; and 65 ml of 0.1 M CaCl₂. The elution pattern can be seen in Fig. 1, and the results are given in Table II.

Of the 4,566,240 units of total input, we recovered 4,052,710 (88%) in the 17 mM CaCl₂ eluate at a mean potency of 1,110 u/A, representing a mean purification factor of 1.97. From this point on in the fractionation calcium was added to all buffers except those used with hydroxylapatite columns because there were inconsistent results and appreciable losses of activity when buffers without calcium were used. For the next step in purification, we selected three eluates from DEAE-agarose columns, amounting to 2,480,400 units (61% of the total yield) with a mean potency of 1,750 u/A.

**Sulfopropyl-Sephardex Chromatography** — the three eluates (17 mM CaCl₂ from DEAE-agarose columns were desalted and concentrated on a UM-10 ultrafilter, then dialyzed against 2 liters of 5 mM CaCl₂, pH 7.5, overnight. In the sample run described below, 30 ml of dialyzed solution were brought to pH 4.50 by dropwise addition of 0.1 M HCl; the small amount of precipitate formed was removed by centrifugation and washed with 5 ml of 5 mM CaCl₂, pH 4.5. The wash, pooled with the supernatant, was applied to a sulfopropyl-Sephardex column (15.0 × 2.5 cm in diameter; bed volume, 78.3 ml) which had been equilibrated with 5 mM CaCl₂, pH 4.50. The absorbance units to bed volume (ml) ratio was 2.47. We found that a low value of this ratio is critical for optimal fractionation on sulfopropyl-Sephardex; for example, if the absorbance unit to bed volume ratio was greater than 10, almost all of the activity was found in the effluent fraction. The following buffers were used in developing the column. Input was: 5 mM calcium acetate, pH 4.50, specific conductivity = 1,075 μmho cm⁻¹. Eluting buffers were: 7.5 mM calcium acetate, pH 4.70, specific conductivity 1,500 μmho cm⁻¹; 12.5 mM calcium acetate, pH 5.25, specific conductivity = 2,100 μmho cm⁻¹; 15 mM calcium acetate, pH 5.5, specific conductivity = 2,400 μmho cm⁻¹; 0.1 M calcium acetate, 0.01 M Tris, pH 7.24, specific conductivity = 11,500 μmho cm⁻¹. The column was run at 0.4 ml/min at 4°C, and 200-drop fractions were collected. After a reading was taken at 278 nm and the appropriate pools were made, the solutions were neutralized (within 1 h after elution), and aliquots were removed for assay and stored at -20°C. The elution pattern is presented in Fig. 2 and results of the fractionation are shown in Table III.

The overall results of this step in the purification were: 55% recovery (1,352,810 units) in the 12.5 mM calcium acetate, pH 5.55 fraction, at a mean potency of 11,170 u/A, and with a mean purification factor of 6.38.

**Gel Filtration** — the 12.5 and 15 mM calcium acetate eluates from the sulfopropyl-Sephardex column separations were run in two separate batches on the same gel column. The pools were concentrated on Amicon UM-2 ultrafilters to about 5 ml and equilibrated with 10 mM CaCl₂, 0.01 M Tris, pH 6.87, before application to the column. The Sephardex G-100 gel was degassed under reduced pressure and equilibrated with the same buffer before the column was prepared. The column (100 × 2.5 cm diameter) was calibrated with markers of known molecular size before being used for the erythropoietin fractions. The void volume was 135 ml; bovine serum albumin monomer eluted at 234 ml, ovalbumin at 258 ml, and cytochrome c at 368 ml. The sample was added to the bottom of the column, as was the buffer which was passed through the column at 21 to 22 ml/
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FIG. 2. Sulfopropyl-Sephadex chromatography of erythropoietin. Buffer changes are indicated by arrows as specified in the text.

TABLE III

| Sulfopropyl-Sephadex fractionation | % recovery of activity | Purification factor |
|-----------------------------------|------------------------|---------------------|
| Input                             | 193 198,160 730 100    |                     |
| 5 mM calcium acetate              | 138 5,610 27 3        |                     |
| 7.5 mM calcium acetate            | 8 16,920 3,170 9 4.3  |                     |
| 12.5 mM calcium acetate           | 12 71,240 9,420 36 12.9|                     |
| 15 mM calcium acetate             | 9 18,600 6,360 9 8.7  |                     |
| 0.1 M calcium acetate             | 19 10,810 3,040 5 4.1  |                     |

h by means of a Mariotte bottle with a 42-cm hydrostatic head. Each fraction collected was 4.1 ml (120 drops), and the following pools were made: I, 0 to 131.2 ml; II, 131.2 to 184.5 ml; III, 184.5 to 205 ml; IV, 205 to 258.3 ml; and V, 258.3 to 328 ml (Fig. 3). The first four pools were concentrated by ultrafiltration and aliquots were assayed. In one of the runs, pools I and II contained 17% and 5% of the absorbance units, respectively, but no detectable activity; pool III contained 32% of the absorbance units and 10% of the input activity, yielding a fraction with a potency of 38,850 u/A; and pool IV contained 10% of the absorbance units and 2% of the biological activity. Pool V was not assayed.

For the combined two gel filtration runs, the yield in pool III (184.5 to 205 ml) was 100%, the mean potency was 39,060 u/A, and the purification factor was 3.04.

Hydroxylapatite Chromatography—Hydroxylapatite was packed under unit gravity into a column (6.1 x 1.5 cm diameter) and washed with 500 ml of water and then with 400 ml of 0.5 mM phosphate buffer, pH 7.1, conductivity = 69 µmho cm⁻¹ (Buffer I), by use of a peristaltic pump which maintained the flow at 0.3 ml/min. After the buffer wash, the length of the column was 3.4 cm and the bed volume was 6.0 ml. The input sample was concentrated and desalted on an Amicon DM-5 ultrafilter by adding water to the concentrate and reconcentrating three times. The final concentrate and the wash of the filter were centrifuged at 6,000 x g for 20 min at 4°. The small insoluble pellet was washed once with 0.5 mM phosphate, pH 7.1, and the wash was added to the supernatant. An aliquot for assay was removed and the remainder (22 ml) was added to the column. The ratio of absorbance units added to bed volume (ml) was 1.82. The input buffer was pumped through the column until the effluent A₃₈₀ was less than 0.005 (149 ml), and the following elution schedule was carried out: Buffer II, 1 mM phosphate (pH 7.1, specific conductivity 131 = µmho cm⁻¹, 70 ml); Buffer III, 2 mM phosphate (pH 6.9, specific conductivity = 270 µmho cm⁻¹, 150 ml (Fraction II)); Buffer III, 2 mM phosphate (pH 6.9, specific conductivity = 270 µmho cm⁻¹, 220 ml (Fractions III A and

FIG. 3. Gel filtration chromatography of erythropoietin. The arrows indicate pools made for assay.

TABLE IV

| Hydroxylapatite chromatography of erythropoietin | % recovery of activity | Purification factor |
|--------------------------------------------------|------------------------|---------------------|
| Input                                            | 11.0 381,480 34,680 100|                     |
| Effluent                                         | 0.35  490 1,400        |                     |
| Fraction II                                      | 0.96 125,480 196,620 32| 3.71                |
| Fraction IIIA                                    | 0.85 79,850 95,940 21 | 2.71                |
| Fraction IIIB                                    | 0.32 22,800 71,350 6  | 2.03                |
| Fraction IV                                      | 0.29  6,290 21,090 2  |                     |
| Fraction V                                       | 7.22 17,680 2,450 5    |                     |
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After the DEAE-agarose step, only 61% of the product was used in further fractionation.

| Step         | Input | Product |
|--------------|-------|---------|
|              | u/A   | u/A     | Yield (%) | Mean purification factor |
| DEAE-cellulose| 7,059,670 | 6,976,170 | 89 | 100 |
| Phenol       | 5,186,890 | 5,115,110 | 110 | 92 |
| Ethanol      | 4,666,240 | 4,052,710 | 1,107 | 89 |
| DEAE-agarose | 2,480,400 | 1,352,810 | 11,170 | 55 |
| Sulfopropyl-Sephadex | 1,259,040 | 1,274,480 | 39,060 | 100 |
| Sephadex G-100 | 1,083,650 | 721,160 | 82,720 | 67 |
| Hydroxylapatite | 5,115,110 | 100 |

The abbreviation used is: SDS, sodium dodecyl sulfate.

When both the native and asialo forms of erythropoietin were run on gels at pH 9, one was fixed, stained, and scanned, and the other was sliced into 1.1-mm slices which were put into 0.5 ml of 0.10% bovine serum albumin, 10 mM CaCl₂ in 0.15 M NaCl, and the hormone was allowed to diffuse out of the gel at 4°C for 18 h. On assay by the in vitro method, we found the biological activity coincident with the single band of stained protein (Fig. 6). In view of our previous finding that native sheep erythropoietin was very poorly fixed to polyacrylamide gels and was largely lost during the staining procedure, we adopted the expedient used earlier for the sheep hormone. Fraction II was iodinated with ¹²⁵I, run on a gel at pH 9 which was then cut into 1.1-mm slices, and counted before and after fixation. The results in Fig. 7 show a single peak of labeled hormone, only a fraction of which was fixed. The iodinated hormone was then run on a gel at pH 6 in order to confirm the apparent homogeneity. It became clear that there was a large degree of aggregation at the lower pH, since only a small amount of the radioiodine could be found in the gel, with the major fraction remaining at the origin. We then used the observation of Kawasaki and Ashwell (10), who found that aggregation of a liver glycoprotein could be reduced by the use of Triton X-100. When both the native and asialo forms of erythropoietin were run on gels in the presence of 0.05% Triton X-100 (Fig. 8), we found for the former a single symmetrical peak and for the these fractions further for evidence of heterogeneity. When these fractions were compared by gel electrophoresis at pH 9, it was clear that there was a small, but significant difference in mobility. Fraction II had a mobility relative to the bromphenol blue tracking dye of 0.49, and the value for Fraction IIIA was 0.52. In spite of our finding of similar potency and molecular size, these two preparations must be considered different. The chemical basis for this difference is now being studied.

When we examined Fractions II and IIIA from the two hydroxylapatite columns by gel electrophoresis in SDS (7.5% gels), we found single bands, each with a relative mobility (with reference to the Pyronin Y band) of 0.50. No detectable difference in mobility, in the presence of SDS, between Fractions II and IIIA could be found. Fig. 5 shows the SDS-gel electrophoretic analysis of each of the most active fractions throughout the purification procedure, and Table V summarizes the seven-step method. The overall purification factor was 929 (calculated from initial and final potencies).

Since the hydroxylapatite Fractions II and IIIA appeared to have a single, identical component on SDS gels, we examined

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Fig. 5. SDS-polyacrylamide electrophoretic analysis of the most active fractions from each step in the purification of human erythropoietin. The gels marked M had serum as marker proteins. 1, DEAE-cellulose eluate; 2, phenol-treated; 3, 90% ethanol precipitate; 4, DEAE-agarose (17 mM Ca²⁺) eluate; 5, sulfopropyl-Sephadex (12.5 mM Ca²⁺) eluate; 6, Sephadex G-100 (pool III); 7, hydroxylapatite Fraction II. The gel concentration was 7.5%.

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Fig. 6. Gel electrophoresis (pH 9) of erythropoietin (hydroxylapatite Fraction II). • represents absorbance of the Coomassie blue-stained material, and O represents biological activity. The gel concentration was 8%. T. D., tracking dye.
Examination of iodinated Fraction II, both native and asialo, on SDS gels (11) showed single, symmetrical peaks (Fig. 9) with no evidence of heterogeneity with respect to size.

We measured the absorbance at 278 nm and at 191 nm, using crystalline bovine serum albumin as a standard and correcting for stray light at 191 nm, and found that A$_{191}$ for erythropoietin is 8.51. Using this value, the mean potency of homogeneous human erythropoietin can be expressed as 70,400 units/mg of protein.

**DISCUSSION**

Espada and Gutnisky (1) isolated a fraction, from urine of patients with anemia due to hookworm, that had a potency of about 8,000 units/mg of protein. They claimed, on the basis of a gel permeation experiment, that this fraction was homogeneous; the poor resolution characteristic of this method of analysis, however, makes it necessary to use additional kinds of information to establish purity. In a subsequent paper, Espada et al. (12) claimed that the same preparation was homogeneous by gel electrophoresis at pH 9, although they pointed out that the stained band was diffuse. In addition, these authors showed an immunodiffusion pattern that was inconclusive with respect to immunological homogeneity. Our finding that human erythropoietin has a minimal potency of 70,400 units/mg of protein suggests that Espada's preparation is either about 11% pure or, if it is homogeneous, is largely in the asialo form that has no activity in vivo.

Our previously reported data (2) indicated that the preparation of sheep plasma erythropoietin, with a potency of 9,200 units, was free of any contaminant except for a small amount of asialoerythropoietin. If this is truly the case, then human urinary erythropoietin is 7 to 8 times more active than the sheep hormone, when assayed by the same method. This may be due to a greater sensitivity of rats to the human than to the sheep hormone, or it may indicate that human urinary erythropoietin is intrinsically more active than sheep plasma erythropoietin.

The appearance of two fractions with the same potency, as a result of hydroxylapatite fractionation, suggests a degree of heterogeneity which is not detected upon electrophoresis in SDS, and which might be accounted for by a small difference in the number of terminal sialic acids or of amide groups, or of both. Our findings of single peaks upon electrophoresis at pH 9, pH 6, and pH 7 in SDS constitute reasonable evidence of homogeneity with respect to charge and molecular size for each of the two fractions. At pH 6 in the presence of Triton X-100, the native and asialo forms are clearly separated (Fig. 9), and we could expect to be able to detect an appreciable amount of the latter mixed with the former. With the exception of the small amount of native erythropoietin found in the asialo preparation, both forms appear to be homogeneous.

Without added surfactant, there is a considerable tendency for native and asialo erythropoietin to aggregate at pH 6 and for the asialo form to aggregate at pH 9, but at pH 7 in the presence of SDS and dithiothreitol, both forms appear to be monomeric. The human asialo hormone has an apparent molecular weight of 34,000 in SDS, whereas the native form has an apparent molecular weight of 39,000. These values contrast with the molecular weight of 41,000 found for sheep plasma asialoerythropoietin by the SDS-gel electrophoretic method and the calculated value of 46,000 for the native form.

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*The molecular weight determined by SDS-gel electrophoresis was the same whether we used 7.5% or 10% gels.*
of the sheep hormone (13). When we studied the sheep hormone, it was clear that the Weber and Osborn method of molecular weight determination (11) by gel electrophoresis in SDS was not accurate for the fully sialylated hormone, possibly because of a substantial contribution by the sialic acids to the net charge. At present, we cannot estimate the molecular size of the native human hormone from that of the asialo form since we do not yet have an accurate estimate of the sialic acid content.

The method of iodination of erythropoietin deserves comment. We found with the sheep hormone that the unmodified iodination method (5), in which chloramine-T is used, caused total loss of biological activity. Following the precedent set by Stanley and Metcalf (14), we used a modification of the method of Stagg et al. (6) in which the iodination was carried out in 45% dimethylsulfoxide to protect methionine residues from oxidation. In contrast to the findings of Stagg et al., who found no loss of gastrin activity, and those of Stanley and Metcalf who found no loss of colony-stimulating activity, we found that the iodination method does cause appreciable inactivation of erythropoietin. For the case of the preparation with 4 iodine atoms/molecule, it can be calculated that less than 2% of the hormone would be noniodinated. This would suggest that the 25% of the biological activity that remained was due to the iodinated derivative. Until a method can be found for preparation of a fully active, labeled hormone, this less active, labeled erythropoietin may still be useful for the study of a number of biological characteristics of the hormone which, until now, have not been amenable to experiment.

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