Supplementary Information

Rapid, Single-Step Protein Encapsulation via Flash NanoPrecipitation

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BSA Recovery

The BSA concentration in nanoparticles was measured with a Bradford assay. To prepare the nanoparticles for the assay, the nanoparticles were dissolved in Acetone, which precipitated the BSA. The samples were centrifuged and decanted. Followed by two more acetone washes. After the final decantation, the BSA precipitate was re-dissolved in water and the BSA concentration was measured with the Bradford assay. To determine the amount of BSA recovered after the acetone washes, a known amount of BSA was prepared in the same manner as the nanoparticles samples and the results from the Bradford assay were compared to the initial loading. We found that 98% of the BSA was recovered after the acetone washes.

Table S1. BSA recovery. BSA recovered in Bradford assay sample preparation.

| Initial Loading (mg) | Recovered BSA (mg) | Percent Recovered (%) |
|----------------------|--------------------|-----------------------|
| 0.90                 | 0.88 ± 0.06        | 98% ± 3%              |

BSA-TA Precipitation

(a) BSA and TA were pipetted together at various ratios into deionized water (pH ~6).
(b) BSA and TA were pipetted together at a ratio of 9:5 by mass. The pH of the reservoir and the BSA solution were varied between pH 2-7. Images were taken 24 hrs after mixing when the precipitate settled.

Figure S1a. Precipitation of BSA and TA - Varying BSA:TA ratio.
Figure S1b. Precipitation of BSA and TA – Varying pH.

PS-b-PEG, BSA, TA Nanoparticles

Figure S2. Nanoparticles formulated with PS-b-PEG (28 mg/mL) and TA (5mg/mL) dissolved in acetone and rapidly mixed with BSA (9mg/mL) dispersed in water via FNP.

PEI Properties

Table S2. Size and zeta potential of PEI dispersed in water.

| Molecular Weight | Diameter (nm) | Zeta Potential (mV) |
|------------------|---------------|---------------------|
| 2kDa PEI         | 140 ± 99      | 13.4 ± 2.5          |
| 10kDa PEI        | 5.1 ± 0.7     | 12.3 ± 3            |
| 750kDa PEI       | 101 ± 7       | 18 ± 2.9            |

PEI, BSA, TA Nanoparticles
**Figure S3.** TEM of BSA/TA nanoparticles stabilized with 750kDa PEI with a 200 micron scale bar. The particles are spherical and the size is consistent with DLS measurements.

**Figure S4.** Effect of PEI molecular weight on nanoparticle formulation.

**Table S3.** Effect of PEI molecular weight on nanoparticle formulation.

| Molecular Weight | Diameter (nm) | PDI      | Zeta Potential (mV) |
|------------------|--------------|----------|---------------------|
| 2kDa PEI         | 753 ± 230    | 0.596 ± 0.090 | -4.2 ± 1.1          |
| 10kDa PEI        | 153 ± 7      | 0.125 ± 0.022 | +15.7 ± 1.0         |
| 750kDa PEI       | 101 ± 3      | 0.274 ± 0.007 | +20.6 ± 1.8         |

**Table S4.** Effect of pH of the BSA stream and PEI reservoir on 750kDa PEI nanoparticles.

| pH of BSA | pH of PEI | Zeta potential (mV) | Diameter (nm) | PDI    |
|-----------|-----------|---------------------|--------------|--------|
| 2         | 10        | +23.2 ± 1.7         | 95 ± 8       | 0.340 ± 0.046 |
| 2         | 10        | +28.7 ± 17.9        | 115 ± 10     | 0.324 ± 0.100 |
| 5         | 10        | +18.9 ± 0.9         | 105 ± 6      | 0.289 ± 0.017 |
| 5         | 10        | +38.7 ± 2.0         | 113 ± 8      | 0.304 ± 0.096 |
| 10        | 10        | +24.0 ± 1.8         | 124 ± 13     | 0.434 ± 0.085 |
Figure S5. DLS of the 10kDa PEI nanoparticles formulated under different pH conditions by (A) varying the pH of the BSA stream and (B) varying the pH of both the BSA stream and the PEI reservoir. Stable particles were formed when the pH of the BSA stream was at or above 5 and the pH of the PEI stream was at or above 10.

Table S5. Effect of pH of the BSA stream and PEI reservoir on 10kDa PEI nanoparticle formulation.

| pH of BSA | pH of PEI | Zeta potential (mV) |
|-----------|-----------|---------------------|
| 2         | 10        | +23.2 ± 1.7         |
| 2         | 10        | +31.3 ± 3.6         |
| 5         | 10        | +15.6 ± 0.4         |
| 5         | 10        | +30.2 ± 2.2         |
| 10        | 10        | +15.7 ± 1.1         |

Table S6a. Varying ratio of stabilizer to core for 750kDa PEI nanoparticles.

| Polymer : Core | Initial | 24 hrs |
|---------------|---------|--------|
|               | Size (nm) | PDI    | Size (nm) | PDI    |
| 3 : 1         | 105 ± 2  | 0.279 ± 0.007 | 100 ± 3  | 0.260 ± 0.002 |
| 2 : 1         | 145 ± 39 | 0.442 ± 0.034 | 1257 ± 80 | 0.873 ± 0.109 |

Table S6b. Varying Total Solids of 750kDa PEI nanoparticles.

| Total Solids (mg/mL) | Zeta Potential (mV) | Initial | 24 hrs |
|----------------------|---------------------|---------|--------|
|                      |                     | Size (nm) | PDI | Size (nm) | PDI |
| 2.8 mg/mL            | +23.7 ± 1.8         | 107 ± 7  | 0.276 ± 0.005 | 107 ± 7  | 0.276 ± 0.005 |
| 5.6 mg/mL            | +19.5 ± 1.2         | 99 ± 2   | 0.270 ± 0.009 | 98 ± 1   | 0.264 ± 0.006 |
| 11.2 mg/mL           | +18.0 ± 0.4         | 101 ± 1  | 0.276 ± 0.009 | 101 ± 1  | 0.271 ± 0.005 |

Table S7. Varying total solids concentration and ratio of BSA to TA by mass for nanoparticles made with 10kDa PEI.

| Polymer : Core | Total solids (mg/mL) | Size (nm) | PDI | Zeta Potential (mV) |
|---------------|----------------------|-----------|-----|---------------------|
| 3 : 1         | 5.6                  | 143 ± 8   | 0.166 ± 0.033 | +14.8 ± 1.1 |
| 3 : 1         | 11.2                 | 319 ± 185 | 0.075 ± 0.051 | +11.8 ± 1.0 |
| 2 : 1         | 4.2                  | 136 ± 42  | 0.357 ± 0.055 | +15.0 ± 1.4 |
Figure S6. DLS of 10kDa PEI nanoparticles after dialysis.

Table S8. Effect of ionic strength on particle stability of 750kDa PEI nanoparticles.

| Salt added   | Concentration (mM) | Ionic Strength (M) | Diameter (nm) | PDI       | Zeta Potential (mV) |
|--------------|--------------------|--------------------|---------------|-----------|---------------------|
| Initial 750kDa PEI | 0                  | 0                  | 107 ± 5       | 0.285 ± 0.004 | 18.5 ± 1.3         |
| NaCl         | 10                 | 0.01               | 89 ± 2        | 0.254 ± 0.007 | 12.7 ± 0.5         |
|              | 30                 | 0.03               | 87 ± 9        | 0.276 ± 0.039 | 15.8 ± 1.2         |
|              | 100                | 0.1                | 91 ± 2        | 0.246 ± 0.004 | 14.8 ± 3.4         |
|              | 300                | 0.3                | 84 ± 5        | 0.297 ± 0.032 | 20.0 ± 4.2         |
| CaCl₂        | 10                 | 0.03               | 84 ± 5        | 0.288 ± 0.028 | 23.1 ± 4.0         |
|              | 100                | 0.3                | 85 ± 4        | 0.306 ± 0.0.024 | 21.2 ± 4.0      |

Table S9. Effect of TA on nanoparticle formulation.

| PEI MW  | Formulated with TA? | Size (nm) | PDI       | Zeta Potential (mV) |
|---------|----------------------|-----------|-----------|---------------------|
| 10kDa   | No                   | 141 ± 13  | 0.196 ± 0.053 | + 16.1 ± 1.4       |
|         | Yes                  | 153 ± 7   | 0.125 ± 0.022 | + 15.7 ± 1.0       |
| 750kDa  | No                   | 90 ± 11   | 0.301 ± 0.064 | + 23.5 ± 2.8       |
|         | Yes                  | 101 ± 3   | 0.274 ± 0.007 | + 20.6 ± 1.8       |