Analysis of the Adaptation Capacity of
Staphylococcus aureus to Commonly Used
Antiseptics by Microplate Laser Nephelometry

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Key Words
Antiseptics \cdot Bacterial adaptation \cdot Microplate laser nephelometry \cdot Mupirocin \cdot Staphylococcus aureus

Abstract
Background: Bacterial colonization and infection are important factors in compromised wound healing. Antiseptics have become an alternative for antimicrobial applications as antibiotic resistance is increasing; they have multiple targets with a broad spectrum of activity. Hence, the risk for developing resistance should be low. However, concerns have been raised that their growing use may result in bacteria that are less susceptible. Methods: The capacity of common antiseptics such as silver nitrate, polihexanide, octenidine, chlorhexidine and polyvinylpyrrolidone (PVP)-iodine to induce adaptation in a Staphylococcus aureus strain was analyzed in vitro using microplate laser nephelometry. S. aureus was repeatedly incubated with the respective half maximal inhibitory concentration (IC\textsubscript{50}) over a time period of 100 days. The influence of the continued treatment was determined by in situ monitoring of changes in the dose-response curves and calculation of the current IC\textsubscript{50} values for the substances tested. Results: During the experiment, S. aureus quickly adapted to high concentrations of the antibiotic mupirocin during repeated treatment. Moreover, a significant increase of the IC\textsubscript{50} for silver nitrate was observed over time. On the other hand, no significant difference was observed for polihexanide or chlorhexidine. While the IC\textsubscript{50} for octenidine was also found to increase significantly, although the change was only marginal, reiterated incubation with PVP-iodine led to a decrease in the IC\textsubscript{50}. Conclusion: Repeated treatment of S. aureus with polihexanide, chlorhexidine, octenidine and PVP-iodine did not trigger bacterial adaptation to these substances.

Introduction
Bacterial colonization and infection are important factors in compromised wound healing. A high bioburden can disrupt the regular healing sequence and result in a chronic inflammatory wound, which leads to cellular dysfunction and biochemical imbalance [1]. Antibiotics are often used to battle infection. However, the widespread application of systemic and topical antibiotics is associated with the emergence of resistant bacterial strains such as methicillin-resistant Staphylococcus aureus (MRSA). Therefore, antiseptics have become a criti-
ical alternative for antimicrobial applications. Increasing numbers of products containing antimicrobial agents are manufactured to enhance the management of microorganisms in the healthcare environment. In addition, they can be found in cosmetics and a number of other products such as sports underwear, shower curtains, door handles, bed linen, trolleys, washing up liquid, etc. Hence, the question has been raised whether this increasing usage should be of genuine concern due to the possible development of antiseptic-resistant germs [2].

There are two major mechanisms of resistance, intrinsic and acquired resistance. An intrinsic nonsusceptibility means that an antiseptic is unable to reach its target site in adequately high amounts to achieve a biocidal effect. This includes, for example, a change in cell wall properties, loss or alteration of porins, surface hydrophobicity, or the presence of efflux pumps [2, 3]. Additionally, physiological (phenotypic) adaptation can modulate the intrinsic resistance, e.g. of cells contained within a biofilm. By contrast, acquired resistance results from genetic changes in a bacterial cell and arises by mutation or physiological (phenotypic) adaptation can modulate the intrinsic resistance, e.g. of cells contained within a biofilm. By contrast, acquired resistance results from genetic changes in a bacterial cell and arises by mutation or by the acquisition of plasmids or transposons [4]. Unlike antibiotics, which act selectively on a specific target, antiseptics have multiple targets with a broad spectrum of activity. Hence, the risk for the development of nonsusceptibility should be low. However, microorganisms can adopt strategies to counter toxic environments [5]. So far, bacterial resistance to organic cationic compounds such as chlorhexidine and quaternary ammonium compounds [4, 6], as well as to silver [7, 8], have been described.

In vitro investigations help to clarify and to highlight the mechanisms of bacterial resistance. It has been shown that bacteria can adapt to harmful agents by decreasing their cytoplasmic concentrations through various mechanisms [2]. In addition, in vitro experiments allow the investigation of the capacity of the antiseptics to induce adaptation in bacteria in a controlled and reproducible manner. Several techniques have been used to ‘train’ bacteria like S. aureus or Escherichia coli to develop resistance to antiseptics, such as disc diffusion, stepwise preparation in broth, repeated exposure to sublethal concentrations, or growth in a basal medium plus an antimicrobial agent [9, 10]. In the present study, S. aureus, which is almost universally present in chronic wounds [11], has been used as a model organism to investigate the adaptation capacity of bacteria to various antiseptics using microplate laser nephelometry (MLN). MLN can be used to monitor the growth of microorganisms by the turbidity of the respective medium. While turbidimetry itself requires relatively high concentrations of particles and obeys Beer’s Law, nephelometry is a direct method of measuring light scattered by particles in solution at a right or forward angle to a laser beam. The most common application of laser-based nephelometry in microplate format is the fully automated solubility screen in high-throughput screening laboratories [12]. Nephelometry is further used in clinical chemistry to determine serum immunoglobulin (IgA, IgG, IgM), complement components (C3, C4), acute phase reactant proteins (CRP, transferring), albumin, and α-1-antitrypsin by protein precipitation, or in organic chemistry to quantify macromolecules, e.g. monitoring of a polymerization reaction. In addition, MLN can be used to study the effect of antimicrobial substances on the growth of microorganisms [13, 14]. Compared to other methods used [9, 15], MLN enables high-throughput screening of several antiseptics and repetition over a prolonged time period, e.g. 100 days. Furthermore, possible changes in the MIC can only be determined for most techniques retrospectively while MLN allows in situ recording of dose-response curves and simultaneous monitoring of alterations in the IC50, a parameter that would be affected if bacteria adapted to the antiseptics. Although disc diffusion methods may offer the possibility of observing changes in situ, it depends on the diffusion capacity of the active agent tested. Large molecules such as polihexanide may have a reduced ability to disperse through the agar compared to small molecules, thus influencing the test outcome. As MLN investigations are performed in solution they do not depend on the diffusion capacities of the substances.

The growth of S. aureus under increasing concentrations of the following antiseptics: polihexanide, octenidine, silver nitrate, chlorhexidine and polyvinylpyrrolidone (PVP)-iodine, was monitored by MLN, and IC50 concentrations were calculated from the dose-response curves. For comparison, the antibiotic mupirocin was also tested in the study. Further, S. aureus was repeatedly incubated with the respective IC50 concentrations over a time period of 100 days. The influence of the continued treatment was determined by the calculation of the current IC50 values for mupirocin and the antiseptics tested.

Materials and Methods

Materials

The following antiseptics have been used in this study: silver nitrate (AgNO3, ACS reagent ≥99%, Sigma), polihexanide (Cosmoci® CQ 20% polyhexamethylene biguanide, PHMB, Arch Chemicals, USA), chlorhexidine (chlorhexidine digluconate solution, 20% in H2O, Sigma), octenidine (Octenisept®, 0.1% octeni-
Table 1. IC₅₀ values of the antiseptics compared to mupirocin before and after treatment of S. aureus

| Substance                  | Concentration range | IC₅₀ (ng/ml) | IC₅₀ after 100 days | Slope |
|----------------------------|---------------------|-------------|---------------------|-------|
| Mupirocin, ng/ml           | 1-500               | 12.5        | 103.6***            | 1.1***|
| Silver nitrate, µg/ml      | 1-40                | 5.3         | 10.8***             | 0.06***|
| Polihexanide, µg/ml        | 0.1–1.0             | 0.53        | 0.48 NS             | <0.001|
| Chlorhexidine, µg/ml       | 0.1–1.0             | 0.56        | 0.63 NS             | <0.001|
| Octenidine, µg/ml          | 0.2–1.2             | 0.51        | 0.63*               | <0.01  |
| PVP-iodine, mg/ml          | 0.25–3              | 0.93        | 0.62*               | −0.003**|

The changes in the IC₅₀ values can be expressed as the slope of the linear regression. NS = Not significant. Differences to the other substances tested: *p < 0.05; **p < 0.01; ***p < 0.001.

dine dihydrochloride, Schülke & Mayr GmbH, Germany), and PVP-iodine complex (Sigma). The antibiotic mupirocin (≥95% HPLC; Sigma) was used for comparison.

The wound dressings Suprasorb® X+PHMB and Suprasorb® A+Ag were allocated by Lohmann & Rauscher GmbH and Co. KG (Germany). Other wound dressings tested were obtained from their manufacturers: Kerlix AMD and Covidien AMD (Tyco Healthcare Ltd., UK), Aquacel Ag (Convatec GmbH, Germany), Acticoat Absorbent (Smith & Nephew Ltd., UK), as well as Urgotul Silver, Urgosorb Silver, and Urgocell Silver (Laboratoires Urgo, France).

S. aureus ATCC6538 was purchased from the DSMZ (Deutsche Sammlung von Mikroorganismen und Zellkulturen, Germany). For the cultivation of bacteria, special peptone and ‘Lab-Lemco’ powder for the preparation of Caso-Bouillon were obtained from Oxoid (UK). Columbia agar plates with 5% sheep blood were purchased from BioMérieux (France). NaCl solution was obtained from Fresenius Kabi Deutschland GmbH (Germany) and Tween 20 was purchased from Roth (Germany).

Microplate Laser Nephelometry
Caso-Bouillon (20 ml) was inoculated with 1–2 colonies of S. aureus grown on Columbia agar plates and incubated for 24 h at 37°C under shaking. The resulting cell suspension was diluted 1:10⁵ in serial steps, yielding a working suspension of approximately 5 × 10⁵ CFU/ml. 100 µl of these suspensions were put in the respective wells of the 96-well microplates containing the prepared antiseptic dilutions as described (for concentration ranges refer to table 1). MLN measurement and calculation of the current IC₅₀ were performed, as explained earlier. Exposure to the respective IC₅₀ concentrations of the antiseptics (refer to table 1 for initial IC₅₀) following this protocol was repeated for 100 days. For each substance tested twelve independent experiments were performed.

Identification of S. aureus by MALDI-TOF Mass Spectrometry
Adapted S. aureus strains were grown on Columbia agar plates for 24 h at 37°C. The plates were then sealed and sent to AnagnosTec GmbH (Germany) for identification. Individual colonies were removed from the plates using a sterile pipette tip and prepared for matrix-assisted laser desorption/ionization and time-of-flight (MALDI-TOF) mass spectrometry as described [16]. All mass spectra were acquired with an Ultraflex II MALDI-TOF/TOF mass spectrometer (Bruker Daltonics, Germany) equipped with an all-solid-state smart beam Nd:YAG laser and operated at 100 Hz in the positive linear mode. Each spectrum was obtained by averaging up to 10,000 laser shots acquired at the minimum laser power necessary for ionization of the samples. Data was evaluated by matching the MALDI-TOF fingerprint mass spectra obtained against reference spectra for S. aureus from the SARAMIS database (AnagnosTec GmbH).

Determination of Antibacterial Activity against Silver-Nitrate-Adapted and Polihexanide-Treated S. aureus
The determination of antimicrobial activity was performed according to the internationally recognized Japanese Industrial Standard (JIS L 1902:2002) testing method for antibacterial activity of textiles [17]. Caso-Bouillon (20 ml) was inoculated either with a silver-nitrate-adapted or a polihexanide-adapted S. aureus strain from this study and cultivated for 24 h at 37°C. For the experiments, 400-µg samples of the wound dressings were incubated with each test microbe (200 µl) for 24 h at 37°C. Samples of polyester material were used as control; they are known not to inhibit microbial growth. For bacteria quantification the incu-
bated samples were extracted in a 10-mL 0.9% NaCl solution with Tween 20. Serial dilutions were plated on Columbia agar plates, incubated for 24 h at 37°C and the colonies counted afterwards. The CFU/mL and the total microbial count of the samples (in CFU) were calculated. The growth reduction compared to the starting value was determined according to [17]:

\[ \text{growth reduction (log CFU)} = \log\left[\frac{\text{24 h MW (CFU) control}}{\text{24 h MW (CFU) sample}}\right] \]

The rating was: no antimicrobial activity = <0.5 log microbial growth reduction; slight antimicrobial activity = 0.5 to 1 log microbial growth reduction; significant antimicrobial activity = >1 to ≤3 log microbial growth reduction; strong antimicrobial activity = >3 log microbial growth reduction.

**Statistics**

All values cited are expressed as means ± standard error. One-way analysis of variance was carried out to determine statistical significances (Microsoft® Excel 2000). Differences were considered statistically significant at a level of p < 0.05.

**Results**

*S. aureus* was incubated with increasing concentrations of mupirocin. Whilst low concentrations did not affect bacterial growth, high concentrations exerted a bacteriostatic or bactericidal effect, discernible by the absence of intensification in turbidity. The IC₅₀ was calculated from the growth curve results according to a logistic fit function (fig. 1b). Similarly, the growth curves for *S. aureus* with the antiseptics, silver nitrate, polihexanide, octenidine, chlorhexidine and PVP-iodine were recorded by MLN and the IC₅₀ values were calculated (table 1).

*S. aureus* was further repeatedly incubated with the respective IC₅₀ concentrations over a time period of 100 days. The effect of this treatment with mupirocin and the antiseptics on *S. aureus* was measured every other day by the determination of the current IC₅₀ using MLN. Figure 2 shows the development of the dose-response curves after 20, 40, 60, 80 and 100 days. *S. aureus* showed a distinct and fast adaptation to the antibiotic mupirocin, resulting in an increase of the IC₅₀ over time (fig. 3a). The IC₅₀ value for silver nitrate was also found to significantly increase during repeated treatment (fig. 3b). On the other hand, polihexanide, octenidine, chlorhexidine and PVP-iodine showed a much lower potency to induce adaptation in *S. aureus* (fig. 3c–f). In fact, repeated treatment with PVP-iodine even resulted in more susceptible germs. The comparison of the initial IC₅₀ values and those after treatment for 100 days is shown in table 1.

The changes in the IC₅₀ values can be expressed as the slope of the linear regression. The antibiotic mupirocin demonstrated a significantly higher slope than the antiseptics tested (table 1), indicating a high risk for the emergence of resistance. From the antiseptics, only sil-
ver nitrate showed a slope that was significantly higher compared to polihexanide, octenidine, chlorhexidine and PVP-iodine. This indicates a distinct risk of developing adaptation to silver when low concentrations are used in a clinical setting. No significant change in the IC\textsubscript{50} values could be noted for polihexanide and chlorhexidine, with slope values less than 0.001. Octenidine showed a statistically significant increase of the IC\textsubscript{50} after 100 days; however, the slope value was less than 0.001. The IC\textsubscript{50} of PVP-iodine significantly decreased during treatment and the slope was −0.003.

To ensure that changes in the behavior of \textit{S. aureus} towards the antibiotic and antiseptics tested really occurred by adaptation and were not due to contamination with other microorganisms during the experiment, the treated \textit{S. aureus} strains were analyzed by MALDI-TOF mass spectrometry. All probes were identified as \textit{S. aureus} species by this method, with an accuracy of 99.9\% (fig. 4). Hence, all changes observed can be attributed to alterations in the adapted \textit{S. aureus} strains.

Antimicrobials are increasingly utilized in wound dressings for the treatment of infected or critically colonized chronic wounds. Several dressings containing silver ions or polihexanide are available for this purpose. A test was done to determine if the in vitro adaptation of \textit{S. aureus} to silver nitrate or treatment with polihexanide affects the antibacterial efficacy of these wound dressings. Hence, silver- and polihexanide-containing dressings were tested for their antimicrobial activity against the native \textit{S. aureus} strain, as well as a silver-nitrate-adapted and a polihexanide-treated \textit{S. aureus} strain from this study. Polihexanide-containing wound dressings achieved a strong reduction of the bacterial growth in the case of the native as well as the polihexanide-treated \textit{S. aureus} strain.
S. aureus – Adaptation to Antiseptics

*aureus* (fig. 5a). Differences in the antibacterial efficacy of the various silver-containing dressings studied were observed. However, all dressings tested showed a similar bactericidal effect against the native and the silver-nitrate-adapted *S. aureus* strain (fig. 5b).

**Discussion**

Antiseptics for wound treatment are chosen by (a) how safe and (b) how effective they are. The safety is determined by the influence on the wound’s healing progress, e.g. if it is delayed or even inhibited. The efficacy is mainly concluded from its microbicidal activity and its ability to decrease wound infection. However, it is also important to test the products used for their potential to induce bacterial resistance or adaptation during prolonged exposure.

Mupirocin (pseudomonic acid A), is a polyketide antibiotic with a narrow spectrum against Gram-positive bacteria, particularly staphylococci, including MRSA. It impairs bacterial protein synthesis by binding to isoleucyl-tRNA synthetase and preventing the incorporation of isoleucine into a growing polypeptide chain. Its extensive use in MRSA patients has been followed by a rapid emergence of mupirocin-resistant strains [18, 19]. Using MLN to detect the adaptation of *S. aureus* to mupirocin, a fast and highly significant increase in the IC$_{50}$ of the antibiotic was observed. This most likely confers a true resistance, although genetic characterization was not performed. Two categories of mupirocin resistance have been described for *S. aureus*. Most isolates with a high-level resistance have obtained plasmid-mediated mupA, which encodes a novel isoleucyl-tRNA synthetase [19]. Low-level resistance to the antibiotic arises by mutation in the native isoleucyl-tRNA synthetase gene *ileS* [2,
Fig. 4. Analysis of untreated *S. aureus* and the treated strains by MALDI-TOF for species confirmation. All probes were identified as *S. aureus* with an accuracy of 99.9%.
20, 21]. It is presumed that the changes observed in this study refer to a low-level resistance.

Unlike the antibiotic mupirocin, silver (Ag⁺) works on multiple components of bacterial cell metabolism. Silver ions react with inorganic compounds, organic acids, proteins, DNA and RNA, killing the microorganisms through the inhibition of cellular respiration, interference with DNA replication and alteration of cellular membrane permeability [8]. The commonly used forms are silver-coated dressings that are more effective at killing a broader range of bacteria than cream-based silver applications and are less irritating than silver nitrate solutions [22]. However, concerns are being raised due to the overuse of silver and the consequent emergence of bacterial resistance. Resistance may develop if the concentration of silver is not high enough to exert a bactericidal effect. As maintaining an adequate concentration of active silver ions over time is a challenge, silver resistance is very possible [8]. Molecular genetics of silver resistance show that it can be conferred by plasmids encoding efflux pumps or silver-binding proteins [7]. However, the probability for the transfer of silver resistance genes seems to be low, unstable and difficult to maintain [23]. Using the MLN method, a slight adaptation of S. aureus to silver nitrate could be detected and confirmed by a significant increase of the IC₅₀. However, the antibacterial activity of silver dressings was not found to be altered. This is in accordance with a study by Loh et al. [24] showing that MRSA isolates carrying silver resistance genes remained susceptible to silver dressings. Hence, differences in the effects of silver-containing dressings against S. aureus probably depend on the silver content present and the delivery form of the active silver ion [22].

It could be shown that the positively charged antiseptics, polihexanide, chlorhexidine and octenidine do not induce adaptation during continuous treatment in vitro. These cationic, surface-active substances are effective against a broad spectrum of microorganisms, including Gram-positive and Gram-negative bacteria [25–28]. They interact with negatively charged molecules in the bacterial cell membranes, resulting in their disruption. It could be shown that polihexanide induces aggregation of acidic lipids in the vicinity of the adsorption site. This changes the membrane permeability and may alter the function of membrane-associated enzymes, causing leakage of cytoplasmatic compounds such as K⁺ [26]. Moreover, the increased permeability of the cell wall allows small molecules like chlorhexidine to penetrate into the bacteria cell and act on targets within the bacteria [27]. Hence, its application for decolonization strategies preventing the transmission of MRSA in intensive care units has been suggested [27]. However, Pseudomonas aeruginosa strains exhibiting raised MIC to chlorhexidine have been observed [29] and MRSA strains with decreased susceptibility have been frequently isolated from clinical settings [30, 31]. While other studies also reported the develop-
ment of resistance towards chlorhexidine [32, 33], this was not found in the present study. Although a slight increase in the IC$_{50}$ was noted, the change was statistically not significant. However, the unsuccessful induction of resistance to chlorhexidine is in accordance with studies by Karen et al. [34] and Suller and Russell [9]. Furthermore, Thomas et al. [29] showed that _P. aeruginosa_ strains exhibiting raised MIC to chlorhexidine were no less sensitive to the agent than the parent strain. Hence, an increase in the MIC for an antiseptic in vitro does not necessarily cause a failure in eradicating the microorganisms in vivo. This is most likely due to the higher concentrations used in practice for disinfection [29].

Studies of Seipp and Körber [35], using the agar diffusion test, and Al-Doorí et al. [36], employing five different MRSA clones, also showed that polihexanide and octenidine do not trigger the acquisition of resistance. It can be assumed that these cationic antiseptics do not cause bacterial adaptation due to their multiple target sites and the electrostatic interactions.

PVP-iodine is a complex of polyvinylpyrrolidone and triiodine ions that is widely used as an antiseptic in trauma and orthopedic surgery. The microbicidal spectrum is broad and only short exposure times are needed [37]. Furthermore, PVP-iodine shows high bactericidal activity against resistant bacteria strains, e.g. MRSA or _Enterococcus faecium_ [37] and it was shown that it does not induce adaptation in bacteria [32]. In fact the results presented here prove that repeated treatment with PVP-iodine even led to a more susceptible _S. aureus_ strain in vitro. It is further effective against biofilm formation [32], even in subinhibitory concentrations [38]. Hence, PVP-iodine could be of interest for new applications in decontamination regimens or in the treatment of infected body cavities and joints [37]. However, its application in wound treatment remains controversial, as it may have potential systemic side effects and exert possible cytotoxic effects leading to impaired wound healing [39, 40].

In conclusion, concerns have been raised that the increasing use of antiseptics may result in bacteria that are less susceptible. MLN presents a valuable tool to investigate bacterial adaptation to antiseptics as it allows high-throughput screening, incubation over a prolonged time period, and in situ monitoring of changes in the dose-response curves. Employing MLN, it could be shown that _S. aureus_ quickly adapts to high concentrations of the antibiotic mupirocin during repeated treatment. This clearly indicates that last-resort antibiotics should be used cautiously to ensure their effect on challenging germs such as MRSA, which are resistant to generally used antibiotics. Commonly used antiseptics showed a much lower potency to induce adaptation in _S. aureus_. Only the IC$_{50}$ for silver nitrate was found to significantly increase during treatment. Although silver-containing wound dressings were still as effective against the silver-adapted _S. aureus_ in this study, the results indicate that an overuse of silver in consumer products may raise future problems. However, polihexanide, chlorhexidine, octenidine and PVP-iodine did not trigger resistance in the microbes, presenting themselves as promising alternatives for antimicrobial treatment in the healthcare environment.

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S. aureus – Adaptation to Antiseptics