Tracing the ‘9th Sulfur’ of the Nitrogenase Cofactor via a Semi-Synthetic Approach

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Abstract

The M-cluster is the [(homocitrate)MoFe7S9C] active site of nitrogenase that is derived from an 8Fe core assembled via coupling and rearrangement of two [Fe4S4] clusters concomitant with the insertion of an interstitial carbon and a ‘9th sulfur’. Combining synthetic [Fe4S4] clusters with an assembly protein template, here we show that sulfite can give rise to the ‘9th sulfur’ that is incorporated in the catalytically important belt region of the cofactor after the radical SAM-dependent carbide insertion and the concurrent 8Fe-core rearrangement have already taken place. Based on the differential reactivity of the formed cluster species, we also propose a new [Fe8S8C] cluster intermediate, the L*-cluster, that is similar to the [Fe8S9C] L- cluster but lacks the ‘9th S’ from sulfite. This work provides a semi-synthetic tool for protein reconstitution that could be widely applicable for the functional analysis of other FeS systems.

Graphical abstract
Nitrogenase plays a key role in the global nitrogen cycle, catalyzing the ambient reduction of nitrogen to ammonia at its active cofactor site.\textsuperscript{1,2} Designated the M-cluster (Fig. 1a, b), the cofactor of the Mo-nitrogenase consists of [MoFe\textsubscript{3}S\textsubscript{3}] and [Fe\textsubscript{4}S\textsubscript{3}] subclusters that are bridged by three μ\textsubscript{2}-‘belt’ sulfides (S\textsuperscript{2−}) and one μ\textsubscript{6}-interstitial carbide (C\textsuperscript{4−}), and it is further coordinated by homocitrate at its Mo end.\textsuperscript{3–6} Assembly of the M-cluster has attracted considerable attention because it is biologically important and chemically unprecedented. Previously, we have shown that the M-cluster is assembled through (i) formation of an L-cluster ([Fe\textsubscript{8}S\textsubscript{9}C]), an 8Fe core of the M-cluster, via coupling and rearrangement of a K-cluster (2x[Fe\textsubscript{4}S\textsubscript{4}]) concomitant with incorporation of an interstitial carbide and a ‘9\textsuperscript{th} sulfur (S)’ on NifB, a radical S-adenosyl-L-methionine (SAM) enzyme; (ii) transformation of the L-cluster into a mature M-cluster via insertion of Mo and homocitrate on NifEN, an assembly scaffold; and (iii) transfer of the M-cluster to its binding site in NifDK, the catalytic component of the Mo nitrogenase (Fig. 1a, b).\textsuperscript{7–12} Further, we have identified early steps along the carbide insertion pathway that involve attachment of the methyl group of SAM to a sulfide atom of the K-cluster, followed by abstraction of a hydrogen atom from this methyl group by a SAM-derived 5′-deoxyadenosyl radical (5′-dA•) and further processing of the resultant carbon-radical species into an interstitial carbide (Fig. 1a, b).\textsuperscript{7–9} While these studies have advanced our understanding of the nitrogenase cofactor assembly, certain crucial details of this process, such as the 4Fe-modular nature of the K-cluster, the origin of the ‘9\textsuperscript{th} S’, and the sequence of events between C and S insertion, have remained unclear. Study of homologously expressed NifB from *Azotobacter vinelandii*, the organism most studied in nitrogenase research, has historically been hampered by an apparent instability and inactivity of the isolated enzyme, limiting its study to a few reports that used either the NifEN-B fusion protein\textsuperscript{7–9} or heterologous expression.\textsuperscript{13,14} Recently, we have established that the NifB protein from *Methanosarcina acetivorans* (designated MaNifB) is a functional homolog of its counterpart from *A. vinelandii* that converts the K-cluster to an L-cluster *in vitro* and subsequently donates the L-cluster to the heterologous biosynthetic machinery of *A. vinelandii* for further maturation into an M-cluster.\textsuperscript{13} Identification of a functional methanogen NifB homolog that can be expressed in *Escherichia coli* circumvents the difficulty of obtaining a stable, active form of the *A. vinelandii* NifB protein and presents a more simplified system than the *A. vinelandii* NifEN-B fusion protein.\textsuperscript{7–9} However, traditional reconstitution of FeS clusters (i.e., using FeCl\textsubscript{3} and Na\textsubscript{2}S\textsubscript{2}) for the as-isolated, cluster-incomplete *MaNifB* cannot directly facilitate the assessment of whether the K-cluster consists of two 4Fe modules; rather, this method often results in attachment of excess sulfide aggregates to the FeS clusters in the protein, making it impossible to trace the incorporation of the ‘9\textsuperscript{th} S’ or uncouple this event from that of carbide insertion.

Here, we use synthetic [Fe\textsubscript{4}S\textsubscript{4}] clusters to reconstitute *MaNifB* and trace the origin of the ‘9\textsuperscript{th} S’, which allows us to define the sequence of events between carbide- and sulfur-insertion during cofactor assembly while establishing a semi-synthetic approach that could be extended to functional analyses of other FeS systems.
Results & Discussion

In search of an alternative FeS reconstitution approach, we first examined whether a water-soluble, synthetic [Fe₄S₄] compound ([Fe₄S₄(SCH₂CH₂OH)₄]²⁻, designated [Fe₄S₄]Syn, Fig. 1c) could be used to reconstitute NifH, the obligate reductase component of Mo-nitrogenase. The water-stable [Fe₄S₄]Syn contains an [Fe₄S₄] core coordinated by ligands (i.e., -SCH₂CH₂OH) that are exchangeable with protein-bound cysteines¹⁵,¹⁶ and, therefore, is a suitable agent for ‘clean’ FeS reconstitutions. Indeed, a dark-brown protein (designated NifH[Fe₄S₄]) could be re-isolated following incubation of the FeS-depleted apo-NifH (designated NifHapo) with [Fe₄S₄]Syn clusters (Supplementary Fig. 1). The [Fe₄S₄] clusters in NifH[Fe₄S₄] and the as-isolated holo-NifH (designated NifHholo) were indistinguishable from each other, both displaying the same characteristic EPR features in the 0, +1 and +2 oxidation states and undergoing the same line-shape changes of EPR spectra upon nucleotide binding (Supplementary Fig. 2). Moreover, NifH[Fe₄S₄] and NifHholo displayed nearly identical catalytic profiles when titrated against increasing amounts of the catalytic NifDK component and almost indistinguishable activities in substrate reductions (Supplementary Fig. 2).

Having established a proof-of-concept for the utility of [Fe₄S₄]Syn clusters in FeS reconstitution, we then generated a [Fe₄S₄]Syn-reconstituted form of MaNifB (designated MaNifB[Fe₄S₄]) by re-isolating the protein from an incubation mixture of apo-MaNifB (designated MaNifBapo) and [Fe₄S₄]Syn clusters. Like the FeCl₃/Na₂S-reconstituted MaNifB,¹³ MaNifB[Fe₄S₄] cleaved SAM into two products: (i) SAH, a product formed upon removal of the methyl group from SAM; and (ii) 5′-dA•, a product formed upon hydrogen abstraction from the SAM-derived methyl group by 5′-dA• (Fig. 2a). Moreover, consistent with the presence of multiple [Fe₄S₄] clusters (i.e., a SAM cluster and a K-cluster), MaNifB[Fe₄S₄] displayed a composite, [Fe₄S₄]¹⁺-specific ½EPR features in the dithionite (DT)-reduced state (Fig. 2b, trace 1) that became EPR-silent upon indigo disulfonate (IDS)-oxidation to the [Fe₄S₄]²⁺ state (Fig. 2b, trace 3). In the presence of SAM, the composite ½EPR signal of MaNifB[Fe₄S₄] underwent changes in the DT-reduced state (Fig. 2b, trace 2) concomitant with the appearance of an 8Fe core-specific g=1.92 signal in the IDS-oxidized state (Fig. 2b, trace 4), which was indicative of transformation of a K-cluster (2x[Fe₃S₃]) into an L-cluster ([Fe₈S₉Cl]) via cluster coupling/rearrangement and carbon/sulfur insertion (Fig. 1a, b).⁷⁻⁹,¹³ Such a transformation was further confirmed by an in vitro cluster maturation assay, wherein the K-cluster on MaNifB[Fe₄S₄] was converted into an L-cluster upon SAM treatment, followed by the L- to M-cluster maturation on NifEN and the subsequent transfer of the M-cluster to NifDK, resulting in a reconstituted NifDK protein that was active in substrate reduction (Fig. 2c). It should be noted that the maturation protein NifEN was homologously expressed in A. vinelandii and that the successful cluster transfer between this protein and the archaenal MaNifB demonstrates an effective strategy to mix and match nitrogenase proteins from different kingdoms.

The observation that an 8Fe L-cluster could be generated from two [Fe₄S₄]Syn clusters on MaNifB[Fe₄S₄] provided direct evidence that the K-cluster consists of a [Fe₄S₄] cluster pair; furthermore, it suggested the possibility of using MaNifB[Fe₄S₄] to investigate the origin of the ‘9th S’ without complications caused by excess sulfide in the FeCl₃/Na₂S-reconstituted
MaNifB. However, DT (S\textsubscript{2}O\textsubscript{4}\textsuperscript{2−}), the non-physiological reductant traditionally used in nitrogenase research, is known to break down into a number of S-based products such as sulfide (S\textsuperscript{2−}), sulfite (SO\textsubscript{3}\textsuperscript{2−}) and sulfate (SO\textsubscript{4}\textsuperscript{2−}),\textsuperscript{17,18} all of which could potentially donate the ‘9\textsuperscript{th} S’ source. To prevent introduction of sulfur species by DT and its breakdown products, we used europium(II) ethyleneglycoltetraacetate (Eu\textsuperscript{II}-EGTA, E\textsubscript{0} = −0.8 V at pH \textsuperscript{8})\textsuperscript{19} as a reductant in the cluster maturation assay to examine the ability of three physiologically relevant sulfur forms—S\textsuperscript{2−}, SO\textsubscript{3}\textsuperscript{2−}, and SO\textsubscript{4}\textsuperscript{2−}—to serve as the source of the ‘9\textsuperscript{th} S’ for the K−to L-cluster conversion on MaNifB\textsuperscript{Fe4S4}. Interestingly, only SO\textsubscript{3}\textsuperscript{2−}-supported cluster conversion on MaNifB\textsuperscript{Fe4S4} in the presence of SAM, resulting in the formation of an L-cluster that could be further matured into an M-cluster to reconstitute NifDK (Supplementary Fig. 3). More excitingly, upon substitution of 35SO\textsubscript{3}\textsuperscript{2−} for unlabeled SO\textsubscript{3}\textsuperscript{2−} in the same incubation mixture, the 35S radiolabel could be detected in the re-purified MaNifB\textsuperscript{Fe4S4} (Fig. 3a, left) and further traced to the L-cluster extracted from MaNifB\textsuperscript{Fe4S4} (Fig. 3a, middle). These experiments provide the very first evidence that a DT breakdown product, SO\textsubscript{3}\textsuperscript{2−}, can serve as the external ‘9\textsuperscript{th} S’ source. Notably, when conducted in the presence of DT, the \textit{in vitro} cluster maturation assays do not require supplementation of SO\textsubscript{3}\textsuperscript{2−} (Fig 2c; Supplementary Fig. 3), which is consistent with an accumulation of SO\textsubscript{3}\textsuperscript{2−} upon decomposition of DT in aqueous solutions.\textsuperscript{17,18} The extracted L-cluster could then be used in a maturation assay in which it was first converted to an M-cluster on the scaffold protein NifEN and subsequently used for the reconstitution of NifDK (Fig. 3a, right). In the absence of SAM, however, no significant retention of the 35S radiolabel was observed when 35SO\textsubscript{3}\textsuperscript{2−} was supplied (Fig. 3a), suggesting that the insertion of the SAM-derived interstitial carbide likely precedes the incorporation of the ‘9\textsuperscript{th} S’.

To further tackle the sequence of events between C and S insertion, we monitored cluster conversion on MaNifB\textsuperscript{Fe4S4} in the presence of SAM alone or SAM plus SO\textsubscript{3}\textsuperscript{2−}. When incubated with SAM, MaNifB\textsuperscript{Fe4S4} displayed the same SAM cleavage pattern (Fig. 3b) and EPR spectroscopic changes (Fig. 3c, d) with or without SO\textsubscript{3}\textsuperscript{2−} treatment. However, quenching those incubation mixtures with acid to release methanethiol—a technique previously used to detect the SAM-derived methyl group attached to a K-cluster S atom\textsuperscript{9}—liberated substantially less methanethiol in the presence of SO\textsubscript{3}\textsuperscript{2−} (Supplementary Fig. S4), suggesting an improved stability and, consequently, a reduced acid-susceptibility of the cluster species in the SO\textsubscript{3}\textsuperscript{2−}-treated MaNifB\textsuperscript{Fe4S4}. Additionally, the SO\textsubscript{3}\textsuperscript{2−}-untreated MaNifB\textsuperscript{Fe4S4} was inactive as an L-cluster donor, whereas the SO\textsubscript{3}\textsuperscript{2−}-treated MaNifB\textsuperscript{Fe4S4} was active in donating the L-cluster for further maturation into an M-cluster (Fig. 3e). The observation of the 8Fe-core-specific g=1.92 signal in SO\textsubscript{3}\textsuperscript{2−}-untreated MaNifB\textsuperscript{Fe4S4} after reaction with SAM (Fig. 3d, trace 2) is particularly intriguing, as it implies that the SAM-dependent carbon insertion and the concurrent 8Fe core rearrangement occur prior to the insertion of the ‘9\textsuperscript{th} S’. Moreover, K-cluster fusion with C in the absence of SO\textsubscript{3}\textsuperscript{2−} leads to the formation of a new intermediate species (designated the L*−cluster) that we tentatively assign as a [Fe\textsubscript{8}S\textsubscript{8}C] cluster that closely resembles the L-cluster in architecture but lacks the ‘9\textsuperscript{th} S’ in the catalytically-important belt region of the cluster (Fig. 4). This so-called ‘vacant’ site of the L*−cluster may be occupied by a putative cysteine thiolate or H\textsubscript{2}O ligand, and studies are underway to assess the exact L*−cluster composition and coordination.
Our proposal that the ‘9th S’ is inserted when the inorganic core structure of the M-cluster is in place is consistent with the facile exchange of the belt S of the M-cluster with selenium (Se)\(^{20}\) or replacement of a belt S with CO\(^{21}\) under turnover conditions (Supplementary Fig. 5). Interestingly, analogous to the utilization of an oxidized form of S (SO\(_3^{2-}\)) for the incorporation of the ‘9th S’, the belt S of the M-cluster can only exchange with the Se atom in selenocyanate (SeCN\(^{-}\)) but not with the more reduced Se in selenide (Se\(^{2-}\)) upon turnover.\(^{20}\) It is possible, therefore, that insertion of the ‘9th S’ involves donation of electrons from the electron-rich cluster to the sulfur source, which explains why a more oxidized S species (SO\(_3^{2-}\))—instead of the fully reduced S (S\(^{2-}\))—is used as a source of the ‘9th S’. In the case of SO\(_4^{2-}\), however, the S atom is unable to directly interact with the Fe atoms; additionally, the cluster may be poised at a certain redox potential that does not accommodate the redox change required for the conversion of SO\(_4^{2-}\) to a ‘9th S’.

Within the cell, sulfite is one of the central hubs of sulfur metabolism (Supplementary Fig. 6).\(^{22–24}\) A survey of the genomes of 13 nitrogen-fixing microbes,\(^{25}\) including those of A. vinelandii and M. acetivorans, revealed the presence of genes encoding enzymes involved in various sulfite-generating routes; most notably, more than 10 of these organisms possess genes encoding the 3’-phosphoadenosine 5’-phosphosulfate sulfotransferase (PAPS transferase), which converts PAPS to sulfite as part of the assimilatory sulfate reduction pathway (Supplementary Fig. 6). This observation provides support for the physiological relevance of sulfite or related compounds to nitrogenase assembly. Although we cannot rule out the participation of reactive physiological sulfur sources other than SO\(_3^{2-}\) such as persulfides for the in vivo ‘9th S’ delivery,\(^{26}\) it should be noted that persulfides are not stable under reducing in vitro conditions. Nevertheless, our identification of shared sulfite-producing enzymes among multiple diazotrophs suggests that these pathways could be operative for supplying the ‘9th S’ source to NifB. While details of these events are yet to be elucidated, identification of sulfite as the in vitro source of ‘9th S’ that is inserted at a late stage of cofactor assembly suggests the possibility to specifically label this belt S for future mechanistic studies of nitrogenase (Supplementary Fig. 5).

Moreover, the methodology of using synthetic metalloclusters as reported herein could be applicable to a wide array of FeS-containing enzyme classes. The \([\text{Fe}_4\text{S}_4]\)\(^{\text{Ssyn}}\) cluster could prove particularly useful for reconstituting other heterologously expressed proteins in the radical SAM superfamily\(^{27}\) as well as other \([\text{Fe}_4\text{S}_4]\)-cluster-containing enzymes associated with important processes such as DNA synthesis and gene regulation.\(^{28–30}\) Strategies to incorporate synthetic cofactors within apoproteins have been successfully employed in the hydrogenase field,\(^{31–33}\) to gain mechanistic and biosynthetic insight. These approaches can also be used to generate artificial enzymes with altered reactivities.\(^{34–36}\)

**Conclusions**

In summary, we have demonstrated the utility of a water stable and soluble \([\text{Fe}_4\text{S}_4]\) cluster to reconstitute the nitrogenase protein NifB for the determination of the L-cluster ‘9th S’ source in vitro. In the process, we have discovered that SO\(_3^{2-}\), and not S\(^{2-}\) or SO\(_4^{2-}\), can donate an S atom for this purpose, and that previous successful studies of cluster maturation on NifB appear to have benefited from the presence of the reductant DT, which can potentially
provide \(\text{SO}_3^{2-}\) to the system. More excitingly, we have identified for the first time a putative \([\text{Fe}_8\text{S}_8\text{C}]\) intermediate, the \(L^*\)-cluster, that can be matured to the \(L\)-cluster upon addition of \(\text{SO}_3^{2-}\). Further investigation of the \(L^*\)-cluster on NifB is an ongoing pursuit.

**Methods**

Methods other than protein reconstitution and \(^{35}\text{S}\)-labeling experiments are included in Supplementary Information.

**Protein Reconstitution**

Purified NiFH\textsubscript{holo} or MaNifB was treated with 20 mM bathophenanthroline disulfonate, an iron chelator, in a buffer containing 5 mM MgATP, 2 mM dithionite (DT; \(\text{Na}_2\text{S}_2\text{O}_4\)), 50 mM Tris-\(\text{HCl}\) (pH 8.0) and 500 mM NaCl, followed by incubation at room temperature for 1 h. The mixture was diluted with a buffer containing 50 mM Tris-\(\text{HCl}\) (pH 8.0) and loaded on a Q Sepharose column (GE Healthcare). Subsequently, the column was washed with a buffer containing 2 mM DT, 50 mM Tris-\(\text{HCl}\) (pH 8.0) and 100 mM NaCl prior to elution of apo-NiFH (designated NiFH\textsubscript{apo}) or apo-MaNifB (designated MaNifB\textsubscript{apo}) with a buffer containing 50 mM Tris-\(\text{HCl}\) (pH 8.0). Reconstitution of NiFH\textsubscript{apo} or MaNifB\textsubscript{apo} with \([\text{Fe}_4\text{S}_4]\)\textsuperscript{\text{Syn}} was carried out by adding a dimethylformamide (DMF) solution of \([\text{Fe}_4\text{S}_4]\)\textsuperscript{\text{Syn}} dropwise to NiFH\textsubscript{apo} or MaNifB\textsubscript{apo} at molar ratios of 1:1 and 5:1, respectively, in a buffer containing 2 mM DT, 20 mM BME, 50 mM Tris-\(\text{HCl}\) (pH 8.0) and 500 mM NaCl, with continuous stirring on ice. After incubation on ice for 1 h, the reaction mixture was diluted with a buffer containing 2 mM DT and 50 mM Tris-\(\text{HCl}\) (pH 8.0) and loaded on a Q Sepharose column. The column was then washed with a buffer containing 2 mM DT, 50 mM Tris-\(\text{HCl}\) and 100 mM NaCl prior to elution of the reconstituted NiFH (designated NiFH\textsubscript{[Fe\textsubscript{4}S\textsubscript{4}]}\textsuperscript{\text{Syn}}) or MaNifB (designated MaNifB\textsubscript{[Fe\textsubscript{4}S\textsubscript{4}]}\textsuperscript{\text{Syn}}) with a buffer containing 2 mM DT, 50 mM Tris-\(\text{HCl}\) (pH 8.0) and 500 mM NaCl. The DT-free reconstitution of MaNifB\textsubscript{apo} was carried out by passing MaNifB\textsubscript{apo} through a Sephadex G-25 desalting column (GE Healthcare) to remove DT, diluting the protein with a buffer containing 20 mM BME, 50 mM Tris-\(\text{HCl}\) (pH 8.0), and treating the protein solution with a DMF solution of \([\text{Fe}_4\text{S}_4]\)\textsuperscript{\text{Syn}} as described above. The mixture was directly loaded on a Q Sepharose column after incubation on ice for 30 min, and the column was washed with a buffer containing 0.5 mM Eu\textsuperscript{II}-EGTA, 50 mM Tris-\(\text{HCl}\) (pH 8.0) and 500 mM NaCl to elution of the reconstituted NiFH\textsuperscript{[Fe\textsubscript{4}S\textsubscript{4}]}\textsuperscript{\text{Syn}} or MaNifB\textsuperscript{[Fe\textsubscript{4}S\textsubscript{4}]}\textsuperscript{\text{Syn}} with a buffer containing 2 mM Eu\textsuperscript{II}-EGTA, 50 mM Tris-\(\text{HCl}\) (pH 8.0) and 500 mM NaCl. All solutions described above contained 10% (v/v) glycerol.

**\(^{35}\text{S}\)-labeling Experiments** —To monitor the event of ‘9th S’ insertion on MaNifB\textsubscript{[Fe\textsubscript{4}S\textsubscript{4}]}\textsuperscript{\text{Syn}} (see Fig. 3a, left), three different reactions were assembled, each containing, in a total volume of 40 μL, 2 mM Eu\textsuperscript{II}-EGTA, 2 mM Na\textsubscript{2}\text{SO}_3, 10% (v/v) glycerol, 50 mM Tris-\(\text{HCl}\) (pH 8.0) and (i) 22 nmol DT-free MaNifB\textsubscript{apo}; (ii) 22 nmol DT-free MaNifB\textsubscript{[Fe\textsubscript{4}S\textsubscript{4}]}\textsuperscript{\text{Syn}}; and (iii) 22 nmol DT-free MaNifB\textsubscript{[Fe\textsubscript{4}S\textsubscript{4}]}\textsuperscript{\text{Syn}} and 5 mM SAM. All reactions were incubated for 30 min at room temperature with intermittent mixing and then run over Ni Sepharose resin (25 μL packed volume; GE Healthcare) pre-equilibrated with a buffer containing 2 mM Eu\textsuperscript{II}-EGTA, 500 mM NaCl, 10% (v/v) glycerol and 25 mM Tris-\(\text{HCl}\) (pH 8.0). The Ni Sepharose resin was washed with 100 μL of the same equilibration buffer before the bound protein was
eluted with 80 μL buffer containing 2 mM Eu\textsuperscript{II}-EGTA, 250 mM imidazole, 500 mM NaCl, 10% (v/v) glycerol and 25 mM Tris-HCl (pH 8.0). The protein eluent was then diluted with 270 μL equilibration buffer and mixed with Ni Sepharose resin (12.5 μL packed volume; GE Healthcare). Subsequently, the protein-bound Ni Sepharose resin was re-suspended in the equilibration buffer and applied directly onto a Whatman Grade 1 qualitative filter paper (GE Healthcare). The blots were dried and exposed to a GE Healthcare Storage Phosphor Screen GP (20 × 25 cm) for 16 h before imaging was performed on a GE Healthcare Typhoon Trio\textsuperscript{+} variable mode imager. To trace the $^{35}$S radiolabel into the ManifB-bound cluster species (see Fig. 3a, middle), reactions ii and iii above were scaled up by 6-fold, followed by extraction of cluster species from these reaction mixtures using a protocol adapted from a previously described procedure.\textsuperscript{8} The extracted clusters were then blotted onto a GE Healthcare Whatman filter paper for imaging as described above.

**Data Availability**

The authors declare that all data supporting the findings of this study are available within the article and its Supplementary Information files and from the corresponding authors upon reasonable request.

**Supplementary Material**

Refer to Web version on PubMed Central for supplementary material.

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Figure 1. Assembly of the M-cluster

(a) The assembly of M-cluster involves SAM-dependent conversion of K-cluster to L-cluster on NifB, transfer of L-cluster to NifEN, maturation of L-cluster on NifEN upon insertion of Mo and homocitrate (HC) by NifH, and transfer of the resultant M-cluster to NifDK. (b) Coupling of the 4Fe units of K-cluster (2x[Fe₄S₄]) into an 8Fe L-cluster ([Fe₈S₉C]) concomitant with insertion of an interstitial carbide and a ‘9th sulfur’, followed by conversion of the L-cluster to a mature M-cluster ([MoFe₇S₉C]) via insertion of Mo and HC. Carbide insertion begins with methyl transfer from SAM to the K-cluster and hydrogen abstraction from the SAM-derived methyl group by 5′-dA• and continues with further deprotonation/dehydrogenation of the carbon-radical until a carbide appears in the center of the L-cluster. (c) The [Fe₄S₄]Syn cluster ([Fe₄S₄(SCH₂CH₂OH)₄]²⁻) with ligands that are exchangeable for protein-bound cysteines. The hydrogen atoms are omitted for clarity. Color code: Fe, orange; S, yellow; Mo, cyan; C, gray; N, blue; O, red.

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Figure 2. Dithionite (DT)-dependent cluster maturation

(a) HPLC elution profiles of (1) SAM, SAH and 5'-dAH standards, (2) SAM incubated with MaNifBapo and DT, and (3) SAM incubated with MaNifB[Fe₄S₄] and DT, showing that contrary to MaNifBapo, MaNifB[Fe₄S₄] was capable of methyltransfer from one SAM molecule and hydrogen abstraction from this methyl group via a 5'-dA• radical derived from another SAM molecule, generating SAH and 5'-dAH as the respective products of these reactions. (b) EPR spectra of DT-reduced (1, 2) or IDS-oxidized (3, 4) MaNifB[Fe₄S₄] incubated without (black) or with (red) SAM, demonstrating the formation of an L-cluster on MaNifB[Fe₄S₄] through a partial disappearance of the composite S=1/2 signal in the DT-reduced state (3) and the concomitant appearance of an L-cluster-specific, g=1.92 signal in the IDS-oxidized state (4). (c) Activities of MaNifB[Fe₄S₄] upon maturation and transfer of its cluster species to NifDK, showing the competence of MaNifB[Fe₄S₄] as an M-cluster donor for NifDK upon cluster maturation. Activity data were obtained from three independent experiments (n=6) and presented as mean±s.d. (c).
Figure 3. Dithionite (DT)-free cluster maturation

(a) Proteins captured on Ni Sepharose resin after incubation of (1) His-tagged MaNiF\(^{\text{apo}}\) with \(35\text{SO}_3^{2-}\), (2) His-tagged MaNiF\([\text{Fe}_4\text{S}_4]\) with \(35\text{SO}_3^{2-}\), and (3) His-tagged MaNiF\([\text{Fe}_4\text{S}_4]\) with SAM and \(35\text{SO}_3^{2-}\) (left); clusters extracted from samples a, 2 and a, 3 (middle); and activities of \(\text{C}_2\text{H}_2\) reduction by clusters extracted from samples a, 2 and a, 3, after maturation and transfer of clusters to NifDK (right). The \(35\text{S}\) label and activity were only detected in SAM/SO\(_3^{2-}\)-treated MaNiF\([\text{Fe}_4\text{S}_4]\), suggesting incorporation of S from SO\(_3^{2-}\) into a fully-assembled L-cluster. (b) HPLC elution profiles of (1) SAM, SAH and 5’-dAH standards, (2) SAM incubated with MaNiF\([\text{Fe}_4\text{S}_4]\), and (3) SAM incubated with MaNiF\([\text{Fe}_4\text{S}_4]\) and SO\(_3^{2-}\), showing the competence of SAM-treated MaNiF\([\text{Fe}_4\text{S}_4]\) in methyltransfer and hydrogen abstraction with or without SO\(_3^{2-}\)-treatment. (c, d) EPR spectra of Eu\(^{II}\)-EGTA-reduced (c) or IDS-oxidized (d) MaNiF\([\text{Fe}_4\text{S}_4]\) incubated without SAM (black), with SAM (red), or with SAM and SO\(_3^{2-}\) (green), suggesting formation of an 8Fe core of the L-cluster in SAM-treated MaNiF\([\text{Fe}_4\text{S}_4]\) with or without SO\(_3^{2-}\). (e) Activities of MaNiF\([\text{Fe}_4\text{S}_4]\) treated with (1) no additive, (2) SAM, and (3) SAM and SO\(_3^{2-}\) upon maturation and transfer of clusters to NifDK, showing the sole competence of SAM/SO\(_3^{2-}\)-treated MaNiF\([\text{Fe}_4\text{S}_4]\) as an M-cluster donor. Activity data were obtained from three independent experiments (\(n=6\)) and presented as mean±s.d. (e).
Figure 4. Refined model of L-cluster assembly on NifB

The radical SAM-dependent carbide insertion and the concurrent 8Fe core rearrangement precede the incorporation of the ‘9th S’ at the catalytically important belt region of the nitrogenase cofactor, resulting in an L*-cluster ([Fe₈S₈C]) that is nearly indistinguishable from the L-cluster ([Fe₈S₉C]) except for the missing ‘9th S’. The encircled black asterisk represents the ‘vacant’ site of the L*-cluster, which may be occupied by a cysteine thiolate or water molecule. Color code: Fe, orange; Mo, cyan; S, yellow; C, gray; N, blue; O, red.