Review

Molecular Hydrogen: Is This a Viable New Treatment for Plants in the UK?

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Abstract: Despite being trialed in other regions of the world, the use of molecular hydrogen (H2) for enhanced plant growth and the postharvest storage of crops has yet to be widely accepted in the UK. The evidence that the treatment of plants and plant products with H2 alleviates plant stress and slows crop senescence continues to grow. Many of these effects appear to be mediated by the alteration of the antioxidant capacity of plant cells. Some effects seem to involve heme oxygenase, whilst the reduction in the prosthetic group Fe3+ is also suggested as a mechanism. Although it is difficult to use as a gaseous treatment in a field setting, the use of hydrogen-rich water (HRW) has the potential to be of significant benefit to agricultural practices. However, the use of H2 in agriculture will only be adopted if the benefits outweigh the production and application costs. HRW is safe and relatively easy to use. If H2 gas or HRW are utilized in other countries for agricultural purposes, it is tempting to suggest that they could also be widely used in the UK in the future, particularly for postharvest storage, thus reducing food waste.

Keywords: heme; heme oxygenase; hydrogen gas; nitric oxide; postharvest; reactive oxygen species; redox; stress responses

1. Introduction

The use of molecular hydrogen (H2) in agriculture is suggested to have a bright future [1,2]. However, its wide use in agricultural practices, including in the UK, has yet to gain traction despite increasing evidence of its benefits (discussed in Sections 3 and 4).

The evidence that H2 has effects on biological systems is growing. This is exemplified in the biomedical field, where H2 research suggests that this diatomic gas may have an important future as a medical therapy [3,4], and there has been a growing body of evidence for its beneficial effects in the biomedical field for some time. H2-based treatments are reported to be advantageous for a range of human diseases, including COVID-19 [5–7], where the inhalation of H2 was the subject of a recent clinical trial [8]. Other conditions for which H2 is shown to have positive effects include neurodegenerative diseases [9,10], such as Parkinson’s Disease [11], as well as rheumatoid arthritis [12], type-2 diabetes [13], metabolic syndrome [14], Hepatitis B [15] and cancer [16]. Therefore, there seems to be little doubt that the treatment of animal cells with H2, including human cells, has significant effects, hence it is being considered as a future therapy [3,4]. As discussed below, much of the discussion regarding how H2 interacts with human and animal cells is translational to plant cells. Therefore, a full understanding of how H2 functions will be beneficial to many aspects of biological sciences, including informing how H2 could be used in agricultural practices. For the adoption of H2-based treatments in agriculture there needs to be a benefit which outweighs the costs. This is not such a relevant issue for treating human conditions...
but, as H$_2$ production for medical use is developed, it may lead to a better cost–benefit ratio, and therefore be viable for the adoption of some plant treatments. Alternatively, if the use of H$_2$ in agriculture demonstrates clear benefits, then the cost of using H$_2$ for medical applications may also be reduced. Critically, the use of H$_2$ in a clinical setting suggests that it is safe for human consumption, which is an important issue if it is to be used to increase the shelf life of food products.

With such a body of evidence reporting that H$_2$ has significant beneficial effects in humans, the positive effects seen in plants are not unexpected. Such effects are mainly seen when plants are exposed to stress, which aligns with the reports in humans, where H$_2$ is being trialed for use in degenerative conditions [9–11] and for numerous diseases [5–7]. In plants, H$_2$ has been suggested as a treatment which should be adopted by agricultural practices [17], although this has not yet happened in the UK. For H$_2$ to be used widely anywhere in the world, the cost of its application must be less than its benefits. For this to be properly assessed, agricultural trials would need to be carefully monitored. Since such trials have yet to take place, it is understandable that there is no urgency for the adoption of H$_2$-based applications in plants on a large scale. Therefore, at present all the data in the literature are from small trials and laboratory-based experiments. Here, we attempt to pool these data together and discuss the pitfalls and potential benefits of H$_2$ use on plants.

On a cellular level, plants can be exposed to H$_2$ by a variety of means. For example, enzymes such as hydrogenases can generate H$_2$ endogenously, as reviewed in [18]. Although not a higher plant, the H$_2$ production by the algae Chlamydomonas reinhardtii is well characterized [19]. The generation of H$_2$ by this alga is so great that it is recommended as a source of H$_2$ for use as a biofuel [20], although some manipulation of the intracellular mechanisms is needed for the H$_2$ production to be sustained [21]. Such a model of cellular generation of H$_2$ is important for understanding the biochemistry of H$_2$ in higher plants, but if the use of this alga is ever commercially viable as a fuel it has the potential to be an exogenous source of H$_2$, which can be exploited in the future for agricultural purposes. Plants may also be exposed to H$_2$ from external sources. This could be from the microflora in the locale of the plant [22] or from certain treatments, which may be useful in agriculture. Here, many of the physiological effects of H$_2$ are discussed, along with the possible cellular mechanisms of action. Finally, the application and use of H$_2$ for enhanced plant growth and crop production is also discussed.

The activities of plant cells are determined by a suite of control mechanisms, including the arrival of signaling molecules from endogenous and exogenous sources. Among these molecules are several which are gaseous in nature and are termed gasotransmitters. These include ethylene [23], nitric oxide (NO) [24] and hydrogen sulfide (H$_2$S) [25,26]; the actions of these compounds are the focus of intense investigation. NO activity has been recognized in plants for more than forty years [27] but, more recently, it was acknowledged that H$_2$S [28] and H$_2$ are also important compounds [29]. However, it is unlikely that such molecules act alone. There are many interactions between these signaling molecules, partly because of their reactivity similarities. Similar to the reactive oxygen species (ROS), both NO and H$_2$S can react with the thiol groups [30,31]. These signaling molecules also seem to modulate the antioxidant capacity of cells [32,33], as well as react with one another. The interactions of ROS, NO and H$_2$S have recently been extensively reviewed [26,34–37]. However, what is pertinent here is that H$_2$ is thought to be part of this web of small, often gaseous, but relatively reactive signaling molecules, and therefore modulating its presence and concentration may have profound and positive effects on plants [17]. H$_2$ is shown to affect the growth and fitness of plants [38] with such favorable effects mediated through the levels of phytohormones [39] such as gibberellin and auxin [40]. As mentioned above, this has parallels with animal studies; H$_2$ was shown to increase the cellular antioxidant capacity in both animals and plants [41,42]. However, in both plants and animals, one of the fundamental conundrums regarding the use of H$_2$ in biological systems is understanding its function. As alluded to previously [43] it is hard to envisage how H$_2$ would be sensed by a classical receptor protein, as it is so small and neutral. H$_2$ is also relatively inert, so it is
unlikely to react directly with amino acids, such as cysteine thiol groups, an action common to ROS, NO and H$_2$S [44]. Certainly, to date, no specific receptor has been identified, and no thiol modification for H$_2$ application has been reported in any cell type studied, either in animals or plants. Therefore, there must be another explanation of how H$_2$ interacts with cellular components if its effects are causational.

Early research suggests that H$_2$ has a selective, reductive activity against ROS and reactive nitrogen species (RNS), especially against the hydroxyl radical (·OH) and peroxynitrite (ONOO$^-$) [45], but this is disputed when the kinetics of the reactions are considered [46,47]. One study suggested that the Fe$^{3+}$ of prosthetic groups could be the target, but limited examples were used, and thus no evidence for this was presented [46]. In plants there are many heme groups involved in respiration and photosynthesis, as well as in ROS and NO metabolism, all which may be potential H$_2$ targets which could account for the effects seen in H$_2$ application [46]. The possibilities of such reactions from a redox potential point of view, with no allowance for kinetics, were recently discussed [48]. It may be possible for some heme prosthetic groups to be targets, but the experimental data need to be gathered to support this view. Alternatively, it was suggested that the hydrogen spin states should also be considered [49].

There is mounting evidence that the antioxidant capacity of cells is altered by H$_2$ administration [41,42], but the direct link back to the molecular targets of H$_2$ has yet to be established. There are reports of heme oxygenase-1 (HO-1) mediating the H$_2$ effects, which account for some of the responses seen in animals [50,51] and plants [52,53]. In fungi, glutathione peroxidase, a selenium-based enzyme, was shown to mediate hydrogen-rich water (HRW) effects [54], although how H$_2$ interacted in this case was not specifically determined.

Regardless of the limited understanding of how H$_2$ functions in cells, there is a growing body of evidence which describes H$_2$ as eliciting positive responses in plants; therefore, its use in agriculture should be considered. The application of molecular hydrogen appears to be safe, leaving no toxic by-products, and has beneficial effects for plants and plant materials such as crops. Undoubtedly, the direct molecular interactions will be understood in time, but this should not be the reason to delay its introduction into agricultural practices.

2. H$_2$ as a Future Plant Treatment

For H$_2$ to be useful for any aspects of agricultural practices it must be safe, cost-effective and easy to apply. Indeed, there are other gaseous treatments, including some of the gasotransmitters discussed above. For instance, NO and donor molecules which release NO are suggested to be useful [55], as is H$_2$S [56], including the use in postharvest storage [57]. However, both NO and H$_2$S are inherently toxic [58,59], whereas there are no reports of H$_2$ toxicity to date, although, to confirm this, clinical trials assessing H$_2$ toxicity are planned [60]. However, as mentioned, the use of H$_2$ in a clinical setting suggests that it is safe, and can be used in food products.

Accordingly, for H$_2$ to have an advantage over other gaseous treatments, it needs to have a more significant effect and be safer and easier to use. The use of H$_2$ also needs to cost less than the amount of enhanced profits that it might elicit. It is hard to envisage a widespread usage for H$_2$ in field agriculture in its gaseous form, partly as it is highly flammable [61,62], and thus a potential safety hazard. However, in the biomedical field, gaseous application is an option as such substances can be directly breathed in, with hydrogen/air, or hydrogen/oxygen mixtures delivered via a nebulizer, for example [63]. It can also be administered at concentrations below the flammability level (<4.6%) and still exert therapeutic effects. Furthermore, hydrogen gas is lighter than air, as exemplified by its use in aircraft [64], so it is difficult to use as a treatment for ground-level plants. In the biomedical field this can be overcome by inhalation [65], but this is not an option for plants unless H$_2$ is used within a closed environment, which would require additional safety protocols. The volume of H$_2$ needed to treat widespread fields would simply not be practical from a production/delivery/application, or even from a cost analysis, point
of view. The use of H$_2$ as a gas may be envisaged on a small scale, such as in some urban settings [66], but again safety considerations would need to be addressed, as would fiscal viability.

To overcome the issue of H$_2$ not being usable as a gas, the most common use of hydrogen in treatments is to create a hydrogen-saturated solution, usually by bubbling the water/media with hydrogen gas, using dedicated apparatus that is either commercially available or in development [67]. H$_2$ has a limited solubility [43,68], but even so, hydrogen-rich water (HRW) can be easily created. However, from studying the literature it is impossible to predict the exact concentration of H$_2$ in such a solution, since a wide range of concentrations are used. HRW can be any solution into which H$_2$ is bubbled, so may be saturated (~0.78 mM), supersaturated, or sub-saturated. Therefore, it is very important that any use of HRW is quantified properly. In the biomedical field, a variation of HRW is hydrogen-rich saline (HRS) [69], but this has limited use in plants, especially in agriculture. It might be useful for salt-stressed experiments [70], but it is hard to envisage its widespread use.

In biomedical use, HRW can be orally ingested, but with plants it is added to the soil as feed water or sprayed directly onto foliage [57]. Additional reports demonstrated the positive effects of H$_2$ when seeds were treated with HRW, with favorable effects during the germination process [71]. For postharvest storage use, flower stems are placed into HRW [72] or fruits are dipped or sprayed [73]. As discussed below, the use of H$_2$-based treatments, such as HRW, may be more useful in floriculture and postharvest storage than in any other aspects of agriculture.

The use of HRW may not be as simple as envisaged. Several studies use HRW in a diluted form. For example, the amounts of 1, 10, 50 and 100% HRW (equivalent to 0.0022 mM, 0.022 mM, 0.11 mM and 0.22 mM H$_2$, respectively) were used in a study of mercury stress alleviation in _Medicago sativa_ [74], mitigating oxidative stress and maintaining redox homeostasis. Likewise, 10% HRW (0.022 mM H$_2$) was used in a study of cadmium stress, with a particular focus on sulfur metabolism [75]. In a study of aluminum toxicity, 75% HRW (0.165 mM H$_2$) was found to have superior effects, including augmented root development and plant growth, when compared with 100% HRW [76]. A study in cut flowers used a range of concentrations (1%, 10%, 50% and 100%; equivalent to 0.0022 mM, 0.022 mM, 0.11 mM and 0.22 mM H$_2$, respectively) and found that 1% HRW showed the best flower quality, with less onset of senescence [77]. The data available suggest that there may be differential effects on different plant species or plant tissues, and it cannot be assumed that 100% HRW is the best treatment to use. The frequency of the applications may also need to be considered, all adding to the complexity and possible expense, of H$_2$-based treatments. Therefore, before the adoption of HRW in any setting, the optimal treatment regime needs to be developed, ensuring that the potential benefits outweigh the costs. To date there has been no wide-scale agricultural use or commercial viability seen in HRW application.

A further variation of the use of HRW is the application of hydrogen nanobubble water (HNW) [78]. This has been used by others to reduce plant stress, for example, in the alleviation of copper toxicity [79]. Nanobubbles, which have a diameter of less than 500 nm, have a high internal pressure and a large surface area. They also have a negatively charged surface. Together, these properties are purported to aid in the dissolving of H$_2$, where the resulting solution is supposedly more stable than HRW, meaning that the H$_2$ treatments can be protracted, rather than having a bolus effect. HNW is created by the generation of H$_2$-containing water, which is then infused into distilled water using an HNW generator [79].

The treatment of biological systems with reactive signaling molecules, such as NO, whether in animals or plants, usually involves the application of a donor molecule [80,81]. However, as well as the inherent toxicity of NO and H$_2$S [58,59], donor molecules can also leave behind unwanted and often toxic by-products [82]. Therefore, there are inherit
difficulties with these methods which will need to be overcome if they are to be proven effective and adopted in the agri-food industry.

As with the other reactive signaling molecules (e.g., H\textsubscript{2}S, NO etc.), the easy and safe use of molecular hydrogen in biological systems in the future may also involve donor molecules. One emerging compound is AB@hMSN \[83\], an ammonia, borane-loaded, hollow, mesoporous silica nanoparticle. Such donors may make the storage and transport of H\textsubscript{2} safer, although, as with other donor molecules used in biological systems, the effects of the donors or their by-products need to be satisfactorily investigated. One such donor molecule is magnesium hydride (MgH\textsubscript{2}), which gives a longer and more sustained H\textsubscript{2} release into the solution \[84\], if used in a special formulation or coating. Even though magnesium is needed by plants, the use of magnesium-based donors would leave magnesium, plus the chemical coating used to passivate the reaction, as by-products. It may also potentially elevate the pH via the production of hydroxide ions with the potential to cause a stress response \[85\]. The problem of the release of by-products by donors is not restricted to H\textsubscript{2} treatments. The NO donor, sodium nitroprusside (SNP), was found to release cyanide, and this was also found to have significant effects \[86\]. This highlights why care must be taken when donor molecules are mooted as an answer to gasotransmitter delivery. The creation of donors will most likely add to the expense of any potential H\textsubscript{2} applications. The creation of HRW is relatively cheap and easy to perform, so any added expense of using a donor molecule will need to be offset by the added biological benefits (better growth, crop yield or postharvest storage), or allow a cheaper treatment regime, perhaps meaning that the application is less frequent, thus saving material and/or labor costs. The future may also see more analogous compounds being developed and sourced from other industries \[87,88\] which could reduce the cost of potential donor products.

For H\textsubscript{2} treatments and applications to be adopted in the UK, strategies for their use need to be put in place. Table 1 summarizes the present options which may be available, once the H\textsubscript{2} has been produced.

| Treatment | Comments | References |
|-----------|----------|------------|
| H\textsubscript{2} as a gas | Safety issues, such as flammability and gas being lighter than air, make it hard to administer. Likely not easy to adopt widely. Useful in biomedical arenas. | \[65,89\] |
| Hydrogen-rich water (HRW) | Likely the preferred method for current plant treatment, either soil feedwater or foliar. Requires optimization for any plant being treated. May require repeated treatments, so adds to expense. | \[70,72\] |
| Hydrogen-rich saline (HRS) | Not useful for agricultural use. Mainly useful for biomedical use, unless considering plant salt stress. | \[69\] |
| Hydrogen nanobubble water (HNW) | Has better H\textsubscript{2} solubility and sustained H\textsubscript{2} release. May require more specialist equipment and expense, although may decrease the regularity of applications needed. | \[78\] |
| Magnesium hydride (MgH\textsubscript{2}) | Potentially provides longer kinetics of action. May leave behind unwanted by-products. | \[84\] |
| AB@hMSN | Long H\textsubscript{2} release. Leaves behind by-products for which the effects need to be determined. Increases the expense of application but might reduce treatment frequencies. | \[83\] |
| Other hydrogen donors | Needs future investment and is cognizant of by-products. ay be an increase in expense, unless widely adopted by other industries. | \[87,88\] |

When any H\textsubscript{2} solution is used, it is important to determine the actual concentration of H\textsubscript{2} present. There are a variety of methods available to do this. Some reports use gas chromatography (GC), e.g., \[90\], whilst others use H\textsubscript{2} m against a standard calibration curve, e.g., \[78\], or an assay based on methylene blue \[91\]. It is only by knowing the actual concentration of the H\textsubscript{2} present that these studies can be compared. Such details must be known before any pragmatic application regime can be determined and costed.
3. H2 and the Alleviation of Plant Stress

Plants, being sessile, are exposed to a range of stress challenges, and this is likely exacerbated by changes in climate, an issue which is predicted to affect the UK [92] and the rest of the world [93]. Improving plant resilience [94], or designing new treatments to mitigate against stress challenges, is important for future food security. The treatments based on molecular hydrogen may be a useful adjunct to other measures in the future. Plants are potentially exposed to a wide range of stresses, including salinity, heavy metals and light, and, since they are sessile, must respond and adapt to these stresses. Although this is an international problem it also applies to the UK, with water quality being a much discussed issue [95]. Therefore, it is important to understand how H2 treatments could help to alleviate any of the potential problems the UK agriculture may face in the future. 

If H2 is to be used in agriculture, there will need to be positive reasons and evidence for its benefits, which outweigh the effort and expense needed for its implementation. As can be seen in Table 2, there is a range of conditions for which H2 can be used to alleviate the negative effects of stress in plants.

Table 2. A selection of plant stresses and the effects of molecular hydrogen.

| Stress          | Comments/Effects Seen on H2 Treatment                                                                 | References |
|-----------------|-------------------------------------------------------------------------------------------------------|------------|
| Salt            | Salt stress is important in coastal regions, especially if there is sea ingress because of climate change. Stress ameliorated; antioxidants increased; Na+ and K+ movements altered. | [70,71,96] |
| Mercury         | Reduced oxidative stress; redox status maintained; better growth.                                      | [74]       |
| Cadmium (Cd)    | Cd uptake reduced; oxidative stress lowered; redox maintained. Ca2+, H2O2, iron-regulated transporter 1 (IRT1) and zinc-regulated transporter protein 2 (ZIP2) are all implicated. | [72,97–100] |
| Aluminum (Al)   | Al uptake reduced. Nitric oxide (NO) accumulation lowered.                                             | [101]      |
| Extreme temperature | Climate change may lead to more extreme temperatures, both high and low, depending on the geographical region. Increased antioxidants, photosynthetic capacity and heat shock protein 70 (HSP70) expression. | [102] |
| UV-A light      | Increased levels of anthocyanin.                                                                       | [103]      |
| UV-B light      | Alterations of antioxidants and flavonoids.                                                           | [104]      |

Plants can suffer from salt stress (NaCl); the connection between climate change and salt stress has been common knowledge for a long time [105]. The salinity levels of arable land are increasing due to irrigation and rising sea levels [106]. Although some treatments, such as the use of selenium [107], have been mooted, the need to engineer plants which are more salt tolerant has been advised [108].

The treatment of plants with H2 is found to be beneficial in saline conditions. The pretreatment with HRW was found to increase the germination in rice when seeds were under salt stress [71]. HRW increased germination from 48.87% to 58.36%. There was also a rise in the germination of HRW application with no salt present (85% to 90.47%). HRW was used at 50% and 100% (0.11 mM and 0.22 mM H2, respectively). Before widespread adoption, the question is whether the cost of setting up and executing HRW treatment in such a scenario is less than the cost of the seed saved. As discussed, only large-scale trials and proper cost analysis will answer this, for any seeds used. On a mechanistic level, both HRW treatments were shown to activate α/β-amylase and to increase antioxidant enzymes, particularly superoxide dismutase (SOD), catalase (CAT) and ascorbate peroxidase (APx). Further work in rice, using two cultivars, showed that HRW treatment, either in foliage or through irrigation, increased the dry weight of the plants but decreased plant height, suggesting an increase in plant productivity [70]. However, the increased crop productivity was assumed here, and not measured, so cannot be used to inform any cost/benefit analysis. When the molecular mechanisms involved in the HRW amelioration of salt stress in barley were examined, it was found that the plants had a greater rate of sodium (Na+) extrusion,
a process mediated by salt overly sensitive (SOS1)-like Na\(^+\)/H\(^+\) exchange proteins. The plant roots also had a better potassium ion (K\(^+\)) retention, so it was concluded that a better Na\(^+\)/K\(^+\) ratio accounted for the positive effects caused by HRW treatment [96].

An increasing body of evidence indicates that the treatment with HRW, is beneficial for plants grown in conditions of excess salinity, but clearly there is much more that needs to be established here, including a much wider range of plants used and whether the increased germination or crop yields warrant the use of any H\(_2\)-based application.

Even though there are environmental regulations in place for industrial pollution in the UK [109], there are still issues where heavy metals accumulate and become a concern [110,111] to agricultural and urban soils [112,113]. Several reports show that H\(_2\)-based treatments may be useful here.

HRW is demonstrated to confer tolerance to mercury in alfalfa (\textit{M. sativa} L. cv. Biaogan) seedlings [74]. Alfalfa is a valuable forage crop, so this would complicate any cost/benefit analysis. However, 10% HRW (0.022 mM) led to relief of the effect of mercury, showing a reduction of approximately 50% of the fresh weight and dry weight of the roots seen in the mercury treatment. This equated to an approximate increase of 10% in growth with the use of HRW when mercury was present. Significantly, the authors state that alfalfa is often grown in areas where heavy metal content is high, and so HRW use in such situations may be worthwhile. On a cellular level, the activities of hydrogen peroxide (H\(_2\)O\(_2\))-removing enzymes were increased and the intracellular glutathione ratios (GSH:GSSG) were restored by HRW. The overall oxidative stress, induced by mercury, was also reduced and the plants were less stunted. Cadmium (Cd) ion stress was also reduced by HRW in Medicago [97]. Ten percent HRW (0.022 mM) led to an increase of 20% of the fresh weight of root material. No effect of 10% HRW was seen in plants which was not stressed with the heavy metals. However, in HRW-treated plants, Cd uptake was inhibited, and again, as with mercury stress, HRW reduced oxidative stress and the cellular redox status was maintained. Similar data were reported for Chinese cabbage (\textit{Brassica campestris} spp. \textit{Chinesis} L.), where HRW increased Cd tolerance, whilst also favorably altering the antioxidant capacity of the plants’ cells [98]. The fresh weight of the plants was increased by 16% with the treatment of 50% HRW (0.11 mM), with root elongation improving by 48%. More recent data show that the movement of calcium ions (Ca\(^{2+}\)) across the plasma membrane and the generation of H\(_2\)O\(_2\) were important in mediating the effects of HRW [99]. Further work in \textit{Brassica chinensis} (Dongfang 2) (Pak choi) and \textit{Arabidopsis thaliana} showed that the transporter proteins, iron-regulated transporter 1 (IRT1) and zinc-regulated transporter protein 2 (ZIP2), were important for mediating HRW actions [100]. Here, 50% HRW (0.11 mM) relieved the effects of 1 \(\mu\)M and 5 \(\mu\)M Cd on the fresh weight of Pak choi roots by about 50%, but there was no effect of Cd toxicity on the overall plant growth, so HRW could not be deemed beneficial in this instance. This would cast into doubt the benefits gained by the expense of the HRW treatment of crops under this particular level of heavy-metal challenge.

Although not classed as a heavy metal, aluminum (Al) ions are toxic to plants [114] and their presence is regularly monitored in the environment in the UK [115,116]. As has been reported in heavy metals, the treatment with HRW can alleviate the stress induced by the presence of this aluminum ions [101]. The uptake of aluminum ions was reduced by HRW treatment, which alleviated the growth inhibition that the aluminum treatment induced. Interestingly, the data showed that the effects of HRW may be mediated by a reduction in nitric oxide (NO) accumulation in the cells. However, even though a pre-treatment with 50% (0.11 mM) HRW partly alleviated the effects of Al on roots, and 10% (0.022 mM) HRW caused an increase in root elongation without any Al stress. when the HRW and Al were added together there seemed to be no effect of the presence of H\(_2\). This casts doubt over the use of this treatment in the field, where the stress is present before H\(_2\) can be used as a treatment.

H\(_2\), administered as HRW, has, therefore, been found to bestow metal ion tolerance to several plant species, and the mechanisms of action are starting to be discovered. However,
the use of HRW for agricultural purposes needs far more research before it can be suggested as commercially viable.

As with other areas of the world [117], the UK temperatures are projected to increase due to climate change [118,119]. The recent Intergovernmental Panel on Climate Change (IPCC) report [120] shows that global temperature rises are inevitable and need to be taken seriously. Therefore, agricultural practices need to help mitigate this for sustained food security, as well as the other stresses. Once again, H$_2$-based treatments may be of use. In a study of heat stress in cucumber, seedlings were treated with 50 or 100% HRW (0.11 mM and 0.22 mM H$_2$, respectively) for seven days, and then subjected to three days of heat. Leaf curling caused by the heat was reduced, but there was no effect on root growth. Therefore, the commercial value of the crop may be increased, but not the overall yield. In line with the other commercially important crops mentioned, the antioxidant capacity was increased in these plants, along with an increased expression of the heat shock protein 70 (HSP70). Therefore, the authors concluded that HRW increased the plants’ tolerance to heat [102]. So, as with other plant stress exposures, HRW, or future H$_2$ treatments, may be useful in UK agriculture by supporting plant responses and enhancing tolerance to rising and fluctuating temperatures [120].

As the climate and atmospheric ozone levels change, the light to which plants are exposed may alter, perhaps through changes in UV irradiation [121,122]. Clearly, the appropriate light intensity and quality is vital for optimal plant growth and crop production. The exposure of Medicago to UV-B radiation increased endogenous H$_2$ production, an action that was mimicked by the treatment with HRW. The effects of H$_2$ appeared to be mediated by alterations in the cellular antioxidant capacity, partly through the modulation of flavonoid levels [104]. Twenty-two types of flavonoids were shown to have their levels altered by HRW [104]. In a study exposing radish (Raphanus sativus L.) to UV-A it was found that the effects were mediated by increases in anthocyanin. Furthermore, it was suggested that plant hormones, mitogen-activated protein kinases (MAPKs) and Ca$^{2+}$ were involved in this mechanism [103]. However, there were no quantification data on the size of the plants, and from the photograph shown there was little effect of HRW on plant size.

It is becoming clear that there are a range of plant stresses for which the use of H$_2$-based treatments may be useful (Table 2). Many of these are relevant to the UK, and many of these stress challenges are only likely to increase as the climate of the UK changes and pernicious, anthropogenic activity continues [120]. However, the evidence for increased growth and crop production is not exhaustive and much more needs to be investigated before a persuasive argument to adopt H$_2$-based treatments in the UK can be made. On the other hand, it is not just plant/crop growth which may be influenced by H$_2$ treatment, but the subsequent storage and transportation of produce may also benefit from H$_2$-based treatments.

4. Postharvest Senescence and the Beneficial Effects of H$_2$

Crops can be treated with H$_2$ postharvest, and this may be one of the most effective ways that H$_2$ could be used in future UK industries. It was estimated that, in the UK, the food supply chain accounts for 13.1 Mt (13,107,000 tones) of food waste, with around 60% of this waste being comprised of cereals and vegetables [123]. Therefore, any improvements to the transport and storage of crops would have a large impact on future food security. To date H$_2$ is shown to be beneficial in several relevant scenarios (Table 3).

As considered above, one of the easiest ways to use H$_2$-based treatments is to create HRW. This can be sprayed onto the surface of fruits and vegetables or, alternatively, the produce can be dipped into HRW. For the floriculture industry, the flower stems can simply be placed directly into the HRW. As HRW is relatively cheap, easy to use, leaves no harmful by-products, and is safe for human health, the use of such treatments in postharvest may have the most potential for future H$_2$ adoption in the UK (Table 3).
Table 3. Examples of the use of H\textsubscript{2} treatments for postharvest of plant materials. Flowers are all cut. H\textsubscript{2} concentrations are calculated from information given in relevant papers.

| Plant Material Being Treated | Treatment | Effects Seen | Reference |
|------------------------------|-----------|--------------|-----------|
| Kiwifruit                    | HRW (80% best: 0.176 mM H\textsubscript{2}) | Increased firmness and less rot. Lower oxidative stress. | [73] |
| Kiwifruit                    | H\textsubscript{2} fumigation | Delayed senescence. Involves ethylene metabolism. | [124] |
| Chinese chive                | 1%, 2% or 3% H\textsubscript{2} fumigation | Reduced oxidative stress. | [125] |
| Hypsizygus marmoreus         | HRW (25% best: 0.2 mM H\textsubscript{2}) | Reduced oxidative stress. | [126] |
| Tomato (Solanum lycopersicum cv.) | HRW (0.195 mM and 0.585 mM H\textsubscript{2}) | Delayed senescence and less nitrite accumulation. Vitamin C retained. | [90] |
| Flowers of lily and rose (Lilium spp. & Rosa hybrida L.) | HRW (0.5% 2.25 \mu M H\textsubscript{2}) and 1% (4.5 \mu M H\textsubscript{2}); lily: 50% (0.225 mM H\textsubscript{2}); rose | Improved vase life. Greater flower diameter. Less oxidative stress. | [72] |
| Lily flowers (Lilium “Manissa”) | HRW (1%: 0.0022 mM H\textsubscript{2} and 150 \mu M sodium nitroprusside (SNP)) | Enhanced flower freshness. ATP synthase CF1 alpha subunit (AtpA) up-regulated. | [127] |
| Lisianthus flowers           | HRW (0.078 mM H\textsubscript{2}) | Vase life extended. Redox maintained as less oxidative stress. | [128] |
| Rose ‘Movie star’ flowers    | HRW (1% best: 0.00235 mM H\textsubscript{2}) | Less flower senescence, mediated by ethylene metabolism. | [77] |
| Carnation flowers            | MgH\textsubscript{2} (0.1 g L\textsuperscript{-1} with citrate) | Flower quality improved. H\textsubscript{2}S signaling downstream of H\textsubscript{2}. | [84] |
| Carnation flowers (Dianthus Caryophyllus L.) | Hydrogen nanobubble water (HNW): 5% best (0.025 mM H\textsubscript{2}) | Less senescence so longer vase life. Lower oxidative stress. | [78] |

A range of HRW concentrations has also been used to treat kiwifruit, with 80\% (0.176 mM H\textsubscript{2}) giving the best results: that is, the fruit had less rot and a better firmness when compared with the other treatments [73]. This benefit was seen over a period of two weeks, which could increase the commercial value of the crop. The lipid peroxidation was reduced, whilst the activity of the endogenous antioxidant, SOD, was increased [60]. H\textsubscript{2} fumigation also decreased kiwifruit senescence, and this appeared to involve alterations in ethylene metabolism [124]. Here, the results showed that the firmness of fruit was increased, with the ripening process being slowed, especially when using a two-stage treatment with 4.5 \mu L L\textsuperscript{-1} H\textsubscript{2}. Interestingly, the damage caused by the pathogen, Phomopsis, was also decreased by H\textsubscript{2}. This suggests that the storage and transport of such fruit may incur reduced losses if H\textsubscript{2} is used. More recently, H\textsubscript{2} was shown to increase the storage time of Chinese chive (Allium tuberosum Rottler ex Spreng) [125] by decreasing oxidative stress in the tissues. Significant positive changes were seen in weight loss, decay index and soluble protein content on H\textsubscript{2} treatment after 4 days, again suggesting that the storage time, and therefore commercial value, of this product would be increased with the application of H\textsubscript{2}. Similar data were reported for the fungus, Hypsizygus marmoreus, where 25% HRW resulted in approximately 25% less weight loss and a higher firmness of the mushrooms. Again, the modulation of antioxidants and the reduction in oxidative stress seemed to be the driving factors behind the efficacious outcomes reported [126]. Other studies in Hypsizygus marmoreus showed that H\textsubscript{2} at 0.8 mM could mitigate against several stresses, including CdCl\textsubscript{2}, NaCl and H\textsubscript{2}O\textsubscript{2}. Again, antioxidants and the lowering of oxidative stress seemed to be important [129], although this study was not carried out postharvest. However, the lower concentrations of H\textsubscript{2} in HRW (0.195 mM and 0.585 mM H\textsubscript{2}) did reduce senescence in postharvest tomato [90], with the firmness being significantly improved for up to 16 days in storage. The nitrite accumulation was also reduced in these fruits. By looking at the relevant enzymes involved in nitrite metabolism, it was found that the enzyme responsible for nitrite generation, i.e., nitrate reductase (NR), was decreased, whilst the enzyme used for nitrite removal to ammonia, i.e., nitrite reductase (NiR), was increased. Vitamin C was also maintained in the tomatoes, and the authors suggested that H\textsubscript{2} would be beneficial for the commercial storage of this fruit. Furthermore, it was suggested that H\textsubscript{2} could be
used for a range of crop storage, with the data showing positive effects in lettuce, eggplant, spinach, apple and carrots, as well as daylily flowers [90].

Several studies showed that H₂ treatment was beneficial for cut flowers (Table 3). In cut lily and rose stems, HRW increased the vase life and flowerhead diameters. The vase life of lilies increased by approximately 2.5 days, whilst for roses HRW led to an increased vase life of 3 days. This could be significant for the commercial value of such flowers. On a cellular level, the stomata size was reduced and the effects were mediated by reduced oxidative stress [72]. Further studies in cut lilies, investigating the crosstalk between H₂ and NO, combined with proteomic analysis, revealed that the effects of increased vase life may be mediated by the expression of the chloroplastic protein ATP synthase CF1 alpha subunit (AtpA) [127].

Treatment with H₂ increased the vase quality of cut lisianthus, an effect that was mediated by the increases in autogenous antioxidants and the subsequent reduction in oxidative stress, thereby maintaining the cellular redox and allowing the flowers to last longer [128]. The treatment with very low H₂ concentrations (0.078 mM H₂) increased the vase life of the flowers by 4 days. In addition, ethylene metabolism was implicated in the increased vase life of cut roses [77], where the maximum effect of H₂ (1% HRW (0.0047 mg L⁻¹) was reported to be a little over 2 days. In cut carnations, an H₂ donor (magnesium hydride) increased the vase life by approximately 2 days and further investigation showed that the gas, H₂S, was found to be involved in mediating this response [84].

H₂ treatment postharvest may not be the only option. In a study of daylily [130], the preharvest application of HRW improved the crop yield (by approximately 10%) and also the flower quality, especially in the chilling stress, which may be useful when considering postharvest transportation and storage.

What is clear from such a collection of data is that treatments with H₂, whether as a gas in solution or from donor molecules, can increase the quality of postharvest plant produce. This was shown for a range of fruits and flowers in which many H₂-based treatments were used. Therefore, more work needs to be carried out to optimize the delivery of H₂ and ensure that it is tailored to the crop being used. This safe and easy-to-use manipulation of postharvest plant tissues has great promise for the future of the food and horticultural industries within the UK and internationally. However, as discussed above, the adoption of any H₂-based treatment comes down to a cost/benefit analysis. Several fruits appear to ripen more slowly and retain their firmness, but, unless the amount and value of the sellable product outweighs the cost of treatment, it may be more commercially effective to simply discard the unwanted crop (or use it for compost or fuel), rather than attempt to prolong its commercial viability. On the flipside of this is whether the crop is plentiful enough to allow such waste. As food security becomes a bigger issue around the globe, all treatments which mitigate food loss may become important, and H₂-based treatments may be a useful adjunct to other applications.

5. Conclusions and the Future

Climate change will continue to be an issue for the UK [92], as well as globally [93,120], and agriculture will be impacted by a range of stress challenges, including those caused by anthropogenic activities [131].

As a mitigating strategy, the use of H₂ as an agricultural treatment may have a future [1,2], although much more evidence needs to be gathered before a good case for its widespread adoption in agricultural practices can be made. The use of H₂ gas may not be easy or pragmatic; however, HRW can be used to encourage healthy germination, as a foliar or as root treatment, although the optimal concentration of HRW needs to be determined for different plant species and applications. An alternate approach is to use donor molecules, which can release H₂ over a longer timeframe, although these must be deemed safe if used on food products. However, any commercial treatment has to be beneficial and a deficiency in crops may be more favorable than the use of costly applications. Large-scale H₂ use
relies on other industries, but it is unlikely that the field agricultural practice in the UK will adopt H\textsubscript{2} in the foreseeable future without the necessary agricultural trials.

Although it is clear that H\textsubscript{2} can be used for the alleviation of a range of stresses, including salinity, heavy metals, temperature and light stresses (Table 2), potentially the most promising use of H\textsubscript{2} treatment is in allaying postharvest senescence (Table 3). As with food production around the globe, there are significant losses in the UK [123]; mitigating these losses would be of tangible value, not only to the economy, but to the food security for the country’s population. The treatment of fruit and flowers with HRW is far easier than spraying a whole field, and will no doubt be seen to be commercially more viable as an option for the use of H\textsubscript{2}-based technologies. It is suggested here that, because HRW is relatively cheap to produce and is easy to use, its adoption in postharvest should be a priority. The effects of HRW at postharvest are evident if plants are treated at preharvest [124]. However, a clear cost analysis would still need to be carried out by those prepared to adopt H\textsubscript{2} applications [132]. As of yet, little data are available on any agricultural cost analysis. It is hard to translate a reported increase in root length or weight, into the subsequent consequences of this for crop yield, especially if it is a foraging crop [74]. Therefore, large-scale trials with all the costs and cost benefits measured for a range of important crops are necessary. Such research would need to be carried out in different regions of the world and under different conditions. Most of the positive effects of H\textsubscript{2} are seen when plants are stressed, and there may be little or no effect if plants are grown under optimal conditions, maintained by other cheaper treatments. However, as H\textsubscript{2} is adopted by other industries [133] it is likely that its generation, storage and transportation will become cheaper, making it more attractive in an agricultural setting. H\textsubscript{2} treatments, as HRW, may be combined with other treatments, such as fertilizer, also reducing the dedicated application costs. Even if the current costs are prohibitive, the use of H\textsubscript{2}-based treatments are likely to be pragmatic in the future, and this may be especially useful for a range of crops at postharvest.

The exact mechanisms of the cellular actions of H\textsubscript{2} in plants, or indeed animals, are still not clear. Despite this dearth of clarity on the direct action of H\textsubscript{2}, it is apparent that H\textsubscript{2}-based treatments are useful to ameliorate stress responses in plants; therefore, the use of molecular hydrogen to enhance agricultural and, particularly, floricultural and horticultural outputs should be considered for wider use in the UK. This may be most important for decreasing food waste and increasing food security.

Author Contributions: J.T.H. wrote the draft of the manuscript; G.R., J.M., A.T. and T.W.L. all authors contributed to and edited the final version. All authors have read and agreed to the published version of the manuscript.

Funding: This research received no external funding.

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: No new data was reported here.

Conflicts of Interest: T.W.L. reports personal fees from medical/academic conferences including travel reimbursement, honoraria, and speaking and consultancy fees from various academic and commercial entities regarding molecular hydrogen. G.R. is funded in part by Water Fuel Engineering who have a commercial interest in hydrogen-producing technologies. All other authors report no conflicts of interest.

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