Malfunctioning of the Iron–Sulfur Cluster Assembly Machinery in *Saccharomyces cerevisiae* Produces Oxidative Stress via an Iron-Dependent Mechanism, Causing Dysfunction in Respiratory Complexes

Mauricio Gomez1, Rocío V. Pérez-Gallardo1, Luis A. Sánchez1, Alma L. Díaz-Pérez1, Christian Cortés-Rojo2, Víctor Meza Carmen1, Alfredo Saavedra-Molina2, Javier Lara-Romero3, Sergio Jiménez-Sandovali4, Francisco Rodríguez4, José S. Rodríguez-Zavala5, Jesús Campos-García1*

1 Lab. Biotecnología Microbiana, Instituto de Investigaciones Químico-Biológicas, Universidad Michoacana de San Nicolás de Hidalgo, Morelia, Michoacán, México, 2 Lab. de Bioquímica, Instituto de Investigaciones Químico-Biológicas, Universidad Michoacana de San Nicolás de Hidalgo, Morelia, Michoacán, México, 3 Facultad de Ingeniería Química, Universidad Michoacana de San Nicolás de Hidalgo, Morelia, Michoacán, México, 4 Centro de Investigación y de Estudios Avanzados del IPN, Unidad Querétaro, Querétaro, México, 5 Departamento de Bioquímica, Instituto Nacional de Cardiología, México, D.F., México

Abstract

Biogenesis and recycling of iron–sulfur (Fe–S) clusters play important roles in the iron homeostasis mechanisms involved in mitochondrial function. In *Saccharomyces cerevisiae*, the Fe–S clusters are assembled into apoproteins by the iron–sulfur cluster machinery (ISC). The aim of the present study was to determine the effects of ISC gene deletion and consequent iron release under oxidative stress conditions on mitochondrial functionality in *S. cerevisiae*. Reactive oxygen species (ROS) generation, caused by H$_2$O$_2$, menadione, or ethanol, was associated with a loss of iron homeostasis and exacerbated by ISC system dysfunction. ISC mutants showed increased free Fe$^{2+}$ content, exacerated by ROS-inducers, causing an increase in ROS, which was decreased by the addition of an iron chelator. Our study suggests that the increment in free Fe$^{2+}$ associated with ROS generation may have originated from mitochondria, probably Fe–S cluster proteins, under both normal and oxidative stress conditions, suggesting that Fe–S cluster anabolism is affected. Raman spectroscopy analysis and immunoblotting indicated that in mitochondria from *SSQ1* and *ISA1* mutants, the content of Fe–S centers was decreased, as was formation of Rieske protein-dependent supercomplex III$_2$IV$_2$, but this was not observed in the iron-deficient *ATX1* and *MRS4* mutants. In addition, the activity of complexes II and IV from the electron transport chain (ETC) was impaired or totally abolished in *SSQ1* and *ISA1* mutants. These results confirm that the ISC system plays important roles in iron homeostasis, ROS stress, and in assembly of supercomplexes III$_2$IV$_2$ and III$_2$IV$_1$, thus affecting the functionality of the respiratory chain.

Introduction

The iron–sulfur centers (Fe–S) are prosthetic groups in many prokaryote and eukaryote enzymes with redox, catalytic, and regulatory functions. These centers are assembled by the iron–sulfur cluster assembly system (ISC), which has been extensively studied and is known to be involved in the incorporation of the Fe–S centers into apoproteins in both bacteria and eukaryotes [1–5]. In eukaryotes, two main systems of Fe–S-protein biogenesis have been described, the cytosol/nucleus (CIA) and mitochondrial (ISC) machineries. Functionally, the CIA machinery depends on the mitochondrial ISC machinery [6–7]. The ISC assembly machinery for maturation of all cellular Fe–S dependent-proteins (mitochondrial, cytosolic, and nuclear) is also involved in iron homeostasis in prokaryotes and eukaryotes [5,7]. In *Saccharomyces cerevisiae*, the mitochondrial ISC machinery is encoded by the genes *NFS1*, *ISU1*, *ISU2*, *ISA1*, *ISA2*, *JAC1*, *SSQ1*, *YAH1*, *GRX5*, and *IBA457* [5,8]. In yeast, the Ssq1 chaperone and the Jac1 J-protein (co-chaperone) function together to assist in the biogenesis of Fe–S centers of Fe–S-dependent proteins. The Fe–S cluster assembly in mitochondria is initiated by cysteine desulfurase (Nfs-Isd11), which obtains a sulfur group from a cysteine and transfers it to the scaffold proteins Isu1 and its redundant Isu2 protein, assisted by Yah1. This interaction also involves frataxin (Yfh1), which acts as an iron donor or activity regulator. ATPase activity in the Ssq1chaperone is stimulated by the J-type co-chaperone Jac1, during the interaction with the scaffold protein Isu1/Isu2 [5–7]. The protein Isu1 is a substrate for both Ssq1 and Jac1, while Jac1 and Isu1 cooperatively stimulate the ATPase...
activity of Ssq1 [3]. The subsequent cluster transference to recipient apoproteins is assisted by glutaredoxin (Grx5) [5,9].

Recently, the participation of the proteins Isa1 and Isa2 in the maturation of mitochondrial apoproteins containing 4Fe–4S clusters such as aconitate, homoaconitate, and lipoic acid synthase was described in S. cerevisiae; this activity is mediated by physical interaction with the Iba57 assembly-protein [5,10–12]. In contrast, Isa1/Isa2 proteins are dispensable for the generation of mitochondrial [2Fe–2S] and cytosolic [4Fe–4S] proteins, because, although the Isa1 and Isa2 proteins are able to bind iron, they are not used as donors for de novo assembly of the [2Fe–2S] cluster on the general Fe–S scaffold proteins, Isa1/Isa2 [5,12]. Upon depletion of the ISC assembly factor Iba57, which specifically interacts with Isa1 and Isa2, or in the absence of the major mitochondrial [4Fe–4S] protein aconitate, iron is accumulated on the Isa proteins, suggesting that the iron bound to the Isa proteins is required for the de novo synthesis of [4Fe–4S] clusters in mitochondria and for their insertion into apoproteins in a reaction mediated by Iba57. Taken together, these findings define Isa1/Isa2 and Iba57 as a specialized, late-acting ISC assembly subsystem that is specifically dedicated to the maturation of mitochondrial [4Fe–4S] proteins [12].

Iron handling by mitochondria during ISC biogenesis must be tightly controlled to avoid a deleterious increase in the concentration of free iron. Ferrous (Fe$^{2+}$) and ferric (Fe$^{3+}$) iron catalyze the formation of the highly reactive hydroxyl radical (OH$^-$) in the presence of H$_2$O$_2$ and O$_2^-$ species through the Haber-Weiss cycle. In mitochondria, these ROS are physiologically produced as by-products of electron transport chain (ETC) activity. Thus, uncontrolled mitochondrial iron homeostasis causes oxidative damage in DNA, lipids, and proteins via the generation of ROS, which in turn further impairs the function of the ETC and leads to cell death [13].

During the dysfunction of the ISC biogenesis system, accumulation of toxic amounts of iron in the mitochondrial matrix occurs due to upregulation of iron transport systems via the activation of the Aft1 transcription factor [14]. Iron accumulation in yeast requires the activation of vascular iron transporters as well as the Mrs3/Mrs4 mitochondrial transporters, which play essential roles in both cell iron homeostasis and heme and Fe–S clusters synthesis by shutting iron into mitochondria [15–17]. The relevance of impaired ISC biogenesis in mitochondrial iron overload and ETC dysfunction is reflected in human diseases like Friedreich’s ataxia, sideroblastic anemia, and ISCU myopathy, whose development has been associated with defects in human genes coding for proteins involved in ISC biogenesis, such as Frataxin, Glrx5, and IscU, respectively [18]. In this regard, yeast has been a powerful tool to elucidate key molecular aspects of the pathogenesis of these diseases because several steps of ISC biogenesis and recycling are conserved between yeast and higher eukaryotes [19].

On the other hand, the antioxidant defenses, iron homeostasis and Fe–S recycling are important mechanisms to restore the function and integrity of the cell under oxidizing conditions. We have recently demonstrated that the deletion of genetic components of the mitochondrial ISC assembly machinery in S. cerevisiae exacerbates ethanol toxicity via increased ROS generation, provoking depletion of the antioxidant response and leading to apoptosis [20]. Given the central role of mitochondria in iron handling, the role of free iron from Fe–S-containing proteins in ROS production, and the fact that ethanol increases mitochondrial ROS generation, it can be hypothesized that excessive free iron exacerbates the toxicity of ethanol and other stressors in mitochondria by disrupting the functionality of the ETC. To test this hypothesis, we have analyzed the relationship between mitochondrial free iron and the levels of ROS generated by stressors, the amount of Fe–S containing proteins, and their effect on respiratory chain functionality in yeast mutants of the ISC system and affected in mitochondrial iron transport.

**Materials and Methods**

**Yeast strains, growth conditions, and survival tests**

The haploid S. cerevisiae BY4741 (Mat a, his3Δ, leu2Δ0, met15Δ0, ura3Δ0) and its KanMX4 interruption gene mutants, ssq1Δ, grx5Δ, isa1Δ, atx1Δ, mrs4Δ, and afg1Δ were obtained from Open Biosystems. Growth tests were carried out using yeast extract peptone dextrose (YPD) culture medium. Tubes or flasks were prepared with 10 or 50 mL of YPD culture medium and added stressor (H$_2$O$_2$, menadione, or ethanol from Sigma), at the indicated concentrations. Culture medium was inoculated with overnight yeast cultures that had reached an optical density of 0.1 at 600 nm (OD$_{600}$nm) and incubated at 30°C with low-speed shaking (50 rpm). Yeast growth (biomass) was spectrophotometrically monitored at OD$_{600}$nm. A survival test was carried out in yeast cultures grown on liquid YPD medium, collected in the late exponential growth phase and then adding ethanol 10% (v/v) and 10 mM 1, 10-phenanthroline, incubating at 30°C with low-speed shaking (50 rpm). Cell survival was determined by Trypan blue staining, and yeast counts were performed using a Neubauer chamber [20].

**Real-time quantification of ROS in S. cerevisiae cultures**

Intracellular ROS in yeast cultures or cell suspensions were determined using oxidant-sensitive, cell-permeant fluorescent probes and fluorescence was quantified by flow cytometry [20]. Cell cultures were grown to the late exponential phase and samples (100 μL) were loaded with the appropriate fluorescent probe. For mitochondrial ROS determination (mit-ROS, mainly H$_2$O$_2$), yeast suspensions were incubated with 5 μg mL$^{-1}$ of dihydroorhodamine 123 (DHR123; Sigma) and for superoxide (O$_2^-$) determination, yeast were incubated with 5 μM mL$^{-1}$ dihydroethidium (DHE, Molecular Probes, Invitrogen), at 30°C for 2 h in the dark. Then, yeast cell samples were taken to 1 mL with PBS buffer (NaCl 137 mM, KCl 2.7 mM, Na$_2$HPO$_4$·2 H$_2$O 8.1 mM, KH$_2$PO$_4$ 1.76 mM, at pH 7.4) and the fluorescence was immediately quantified by flow cytometry using a BD Accuri C6 Flow Cytometer (BD Biosciences). The populations of cells for each of the treatments were gated in the forward scatter and side scatter dot plots to eliminate dead cells and cell debris. Populations corresponding to auto- or basal-fluorescence were located in the left quadrant and cells with emission of fluorescence increments of at least one log unit value were located in the right quadrant of the dot plots. In addition, the percentage of fluorescent cells (PFC) and the median fluorescence intensity (FI) were determined using the BD Accuri C6 Flow Cytometer (BD Biosciences). The populations of cells for each of the treatments were gated in the forward scatter and side scatter dot plots to eliminate dead cells and cell debris. Populations corresponding to auto- or basal-fluorescence were located in the left quadrant and cells with emission of fluorescence increments of at least one log unit value were located in the right quadrant of the dot plots. In addition, the percentage of fluorescent cells (PFC) and the median fluorescence intensity (FI) were determined using the monoparametric histograms of fluorescence emission obtained from the dot plots and labeled as percentage of cells and as relative units of fluorescence. The equipment was calibrated using Spherotech 8-peak (FL1-FL3) and 6-peak (FL4) validation beads (BD Accuri). Fluorescence of the DHR123 probe was monitored in the emission fluorescence channel FL1 (533/30 nm), and for the DHE probe, in the FL2 channel (587/40 nm). A minimum of 20,000 cellular events were analyzed for each determination point. For stressor treatments and Fe$^{2+}$ dose-response assays, yeast cultures grown on YPD medium (10 mL) were loaded with the fluorescent probes by incubating for 30–60 min, washed with PBS and supplemented with the respective concentrations of ROS-generator compounds or Fe$^{2+}$ solution [FeSO$_4$·(NH$_4$)$_2$ with an equimolar amount of citric acid, Sigma]. At the respective times,
samples (100 μL) were harvested, washed and suspended in PBS, adjusting the volume to 1 mL or 1×10⁷ cells mL⁻¹ and the fluorescence was determined by flow cytometry.

Real-time quantification of Fe²⁺

Iron in the yeast suspensions was determined using the fluorescent, cell-permeable indicator for heavy metal Phen green FL (PGFL; Molecular Probes, Invitrogen), which can be used to detect a broad range of ions, including Cu²⁺, Cu⁺, Fe²⁺, Hg²⁺, Pb²⁺, Cd²⁺, Zn²⁺, and Ni²⁺. Fluorescence of PGFL disappears after binding of free Fe²⁺. Therefore, once cells were loaded with the probe, Fe²⁺ was detected by the addition of 1 mM of the chelator 1,10-Phenanthroline. This treatment leads to PGFL-Fe³⁺ complex dissociation, producing fluorescence [21]. Yeast cells suspensions (1×10⁷ cells/mL) were incubated with PGFL (5 μg/ mL) at 30°C for 2 h in darkness. Then, yeast cells were harvested, washed once, and re-suspended in PBS. Fe²⁺ quantification in yeast suspensions was performed without and with ROS-generator treatment and fluorescence was quantified by flow cytometry monitoring the emission fluorescence in channel FL1 (533/30 nm).

Confocal microscopy of yeast suspensions

Saccharomyces cerevisiae YPD-grown cultures were harvested and suspended in PBS at 1×10⁷ cell/mL and loaded with the fluorescent probe DHE or PGFL as detailed above, incubating with light shaking in darkness. Suspensions were treated with and without ethanol (10%) and incubated for 30 min at 30°C. Afterwards, the cell suspensions were incubated with Rhodamine 123 (Rho123; Sigma) during 30 min for mitochondrial colocalization and analyzed using a confocal microscope (Olympus FV1000). The signal evaluating fluorescence emission was observed between 560–580 nm for DHE, between 405–505 nm for PGFL and between 390–600 nm for Rho123. Images were acquired with different magnifications.

Mitochondria isolation

For the determination of mitochondrial complexes activity, mitochondria of S. cerevisiae were isolated from cultures grown in liquid medium YPD at 30°C in a shaking incubator, using a previously described method with light modifications [22]. Lytic case from Arthrobacter luteus (Sigma-Aldrich) was used instead of zymolyase. Yeast cells were harvested in late exponential growth liquid medium YPD at 30°C in a shaking incubator, using a previously described method with light modifications [22]. Lytic case from Arthrobacter luteus (Sigma-Aldrich) was used instead of zymolyase. Yeast cells were harvested in late exponential growth.
for 5 min with 0.75 mM KH2PO4 buffer incubating for 5 min with 0.75 μM KCN. Basal absorbance at 550 nm during 1 min was registered, and then the reaction was started by adding 10 mM D-lactic acid. Then, the changes in the absorbance were recorded for 5 min. The rate of cytochrome c reduction was determined from the slopes of the absorbance plots using a molar extinction coefficient for cytochrome c of 19.1 mM⁻¹ cm⁻¹ [22].

Determination of lactate-cytochrome c oxidoreductase. This activity was measured in 1 mg/mL freeze-thawed mitochondria re-suspended in 50 mM KH2PO4 buffer incubating for 5 min with 0.75 μM KCN. Basal absorbance at 550 nm during 1 min was registered, and then the reaction was started by adding 10 mM D-lactic acid. Then, the changes in the absorbance were recorded for 5 min. The rate of cytochrome c reduction was determined from the slopes of the absorbance plots using a molar extinction coefficient for cytochrome c of 19.1 mM⁻¹ cm⁻¹.

Determination of cis-aconitase activity

Aconitase activity was determined as described by Henson and Cleland (1967) [25]: 100 μg of mitochondrial protein was suspended in lysis buffer (Tris-HCl 50 mM, Triton X-100 0.02%, pH 7.4) with vigorous shaking and incubated for 5 min. Extracts were centrifuged at 9,900 Xg for 5 min at 4°C, and supernatants obtained were used for cis-aconitase determinations. Eighty micrograms of protein from the supernatants were used for enzymatic determination in reaction buffer (Tris-HCl 90 mM, isocitrate 20 mM, pH 7.4) with lighagitation; the absorbance at 240 nm was immediately recorded.

Mitochondrial membrane potential

Membrane potential in the mitochondrial suspensions was determined using the fluorescent, cell-permeable indicator Rh123. Mitochondrial suspensions (1 x10⁷ mitochondria/mL) were loaded with Rh123 (5 μg/mL) and incubated at 30°C for 30 min in darkness. Suspensions were harvested, washed once and re-suspended in PBS. Membrane potential in suspensions was determined by fluorescence generation and quantified by flow cytometry, monitoring the emission fluorescence in channel FL1 (533/30 nm).

Raman spectroscopy of mitochondria

Suspensions of intact mitochondria (250 μg) from yeast cultures grown on YPD were subjected to Raman spectroscopy. Raman analysis was performed using a microRaman spectrometer (Dilor model LabRam) equipped with a confocal microscope with 50× amplification, using He-Ne laser emitting at 632.8 nm and a 256×64 pixel charge-coupled device (CCD) as a photon detector. The spectra showed correspond to the average of spectra overlapped by 60 s of recording; measurements were carried out at room temperature with sample preparation as described elsewhere [26].

Native Gel Electrophoresis and western blot

For Blue Native gel electrophoresis, samples of 100 μg of mitochondrial protein were solubilized using buffer A containing dodecylmaltoside (1 g/g), triton X-100 (2.4 g/g), and digitonin (3 g/g) as described [27], and separated by native polyacrylamide gel electrophoresis on 8% Bis-tris gels (BN-PAGE); mitochondrial complexes were identified as described [28–30]. For immunodetection assay 50 μg of mitochondrial protein were run in SDS-PAGE 12% polyacrylamide gels and transferred to polyvinylidene difluoride (PVDF) membranes. Membranes were blocked using dry milk in PBS-T and blotted with the S. cerevisiae anti-Rp1p antibody as first antibody in blocking medium at a 1:20000 dilution for 2 h at 4°C [4]; after washing, the membrane was incubated with the secondary antibody, a monoclonal anti-mouse IgG HRP-conjugate (Promega), in blocking medium at a 1:5000 dilution for 2 h at 4°C; the membrane was washed with PBS-T and developed using Supersignal West Pico Luminol (Pierce) and exposing in light-sensitive films. Assays were conducted by triplicate and representative images are shown. Bands intensities in gels or films were quantified using the Image J software.

Results

ROS susceptibility of S. cerevisiae is exacerbated by mutations in the ISC

To verify whether the dysfunction in the ISC system is related to a parallel increase in the sensitivity to oxidative damage, the susceptibility to several ROS-generating compounds was tested using three ISC mutants, whose disrupted genes encode proteins that play important roles in assembly of Fe–S centers. In addition, control S. cerevisiae strains that display a severe imbalance in iron homeostasis were used; these included atx1Δ mutants, which are impaired in high affinity iron-depleted medium, as this gene is involved in copper trafficking and delivery to Fet3p, which oxidizes Fe²⁺ to Fe³⁺ for uptake by Ftr1p [31] and mrs4Δ mutants, which show cellular iron accumulation and sensitivity to H₂O₂ and menadione, as this gene is co-regulated with the iron regulon, and encodes the Msr1 Fe²⁺ iron transporter at the inner mitochondrial membrane under conditions of iron deprivation [15–17]. atf1Δ mutants show increased ROS sensitivity and iron accumulation by inducing iron-sensing genes under iron deprivation conditions [32]. All these yeast strains show iron-dependence when grown in the presence of phenanthroline, which induces iron-depletion, and which was improved with iron addition in the culture media (Figure S1). atf1Δ mutant is more sensitive to iron depletion than mrs4Δ mutant, which in turn correlates with the exacerbated sensitivity of the atf1Δ mutant to ROS inducers (Fig. 1). Of the ISC mutants, sqs1Δ, grx5Δ, and isal1Δ mutants showed a significantly impaired growth rate compared to WT, displaying severely compromised growth at concentrations in the range of 6.25–12.5 mM H₂O₂, grx5Δ mutants were the least sensitive at all concentration of H₂O₂ (Fig. 1b–c). The susceptibility to menadione (a superoxide generator), followed a similar pattern to that observed with H₂O₂ treatments: at 80 μM menadione, ISC mutants showed a moderate but significant inhibition in their growth kinetics, with respect to the WT strain (Fig. 1d), while at 150 μM menadione, also with ethanol (8%), the growth of all ISC mutants was drastically affected (Fig. 1e–f); again, grx5Δ was the least sensitive ISC mutant to the stressor. As expected, atx1Δ and mrs4Δ mutants (which show imbalanced iron homeostasis), showed similar behavior to the WT under iron sufficiency, except in YPD supplemented with 12.5 mM H₂O₂ (Fig. 1c), in which delayed growth was observed. In contrast, the hypersensitive atf1Δ mutant exhibited a markedly sensitivity to H₂O₂, menadione, and ethanol treatments (Fig. 1b–f).

Increment in ROS generation correlates with a dysfunctional ISC assembly system

To elucidate whether the enhanced sensitivity to ROS generators was correlated with increased mitochondrial ROS generation due to ISC mutations, real-time quantification of ROS was performed by flow cytometry, using the fluorescent ROS indicators DHE and DHR1123 to detect mitochondrial O₂⁻ and H₂O₂, respectively. ROS generation was determined as the percentage of fluorescent cells (PFC), corresponding to cells that produced ROS level increments of at least one log unit. All mutants displayed a significant increment in the PFC generating O₂⁻ or H₂O₂ when ROS induction was conducted using H₂O₂.
menadione, or ethanol (Fig. 2, continuous lines), compared to untreated strains (Fig. 2, dotted lines). PFC values in yeast suspensions without ROS-inducers were approximately 10%, while in treatments with ROS-inducers the PFC increased to 40–90%. Moreover, ssq1Δ, isa1Δ, and grx5Δ mutants showed higher PFC values than the WT in all ROS-inducer treatments (Fig. 2), indicating that ISC mutations caused increased ROS generation compared to WT yeast. Importantly, ethanol treatment exacerbated H2O2 generation in ISC mutants, but decreased H2O2 generation in atx1Δ, mrs4Δ, and aft1Δ iron-transport mutants, compared to ISC mutants (Fig. 2f). In addition, atx1Δ, mrs4Δ, and aft1Δ mutants exhibited higher levels of superoxide than H2O2 (Fig. 2c and 2f). grx5Δ mutants, concordant with their lowest sensitivity to oxidants among ISC mutants (Fig. 1), also produced the lowest amounts of ROS, except O2−, under treatment with 10% ethanol (Fig. 2). In addition, O2− production in aft1Δ mutants was higher than in all other ISC mutants, concordant with its higher sensitivity to stressors (Fig. 1). Interestingly, when ethanol was used as the ROS generator, the ISC mutants produced higher levels of H2O2 than the iron-transport mutants, (Fig. 2f). These results indicated that the O2− and H2O2 generation are increased by treatment with ROS inducers in S. cerevisiae and are enhanced by the ISC mutation in a time-dependent manner, as in the defective iron-transport mutants atx1Δ, mrs4Δ, and aft1Δ.

Increment in ROS generation correlates with Fe2+ release and is increased by ISC mutations

To evaluate whether the increment in ROS following treatment with oxidant agents was correlated with the functionality of ISC assembly system and the level of the free iron pool, in vivo real-time free Fe2+ quantification was performed using flow cytometry. Cell suspensions of ssq1Δ, grx5Δ, and isa1Δ mutants showed a significant, time-dependent increment in levels of free Fe2+ compared to the WT strain, independent of the presence or absence of oxidants (Fig. 3). As expected, yeast suspensions treated with toxic concentrations of H2O2 (12.5 mM), menadione (80 μM), or ethanol (10%) exhibited higher iron- and time-dependent fluorescence increments than untreated yeast strains, leading to a 2–5 fold augmentation of the quantity of free Fe2+ in ISC mutants after 6 h of treatment. Notably, ssq1Δ and isa1Δ mutants showed higher levels of Fe2+ than grx5Δ mutants. In addition, Fe2+ release was higher in atx1Δ mutants than in aft1Δ and mrs4Δ mutants, and in these mutants, free Fe2+ was also increased when they were treated with oxidant agents (Fig. 3). However, aft1Δ and mrs4Δ mutants showed behavior intermediate between the WT strain and ISC mutants; these cells showed increased Fe2+ than WT, but lower free Fe2+ than ISC and atx1Δ mutants (Fig. 3b–c). When the Fe2+ release data were analyzed following 6 h of treatment with ethanol, a significant increment in fluorescence values was observed for ssq1Δ, grx5Δ, and isa1Δ strains, but not for the iron-transport defective strains atx1Δ, mrs4Δ, and aft1Δ. However, in the absence of ethanol, only ssq1Δ and isa1Δ mutants showed significant differences in free Fe2+ release values (Fig. 3d). As mentioned above, with ethanol treatment, the free Fe2+ value was significantly increased in all strains, compared to untreated yeast cultures.

In addition, microscopic analysis showed that ethanol treatment caused an increment in superoxide generation, associated with release of free Fe2+ and being exacerbated in ISC mutants (Fig. 4). For example, in both WT cells and ssq1Δ mutants, O2− levels were higher under treatment with ethanol than with glucose; this
response was exacerbated in ssq1Δ mutants, consistent with iron-fluorescence levels determined using the PGFL probe, indicating that it was associated with free Fe2+ release (Fig. 4a–h). Interestingly, in the WT strain grown on glucose, O2− production and free Fe2+ were co-localized in mitochondrial structures (Fig. 4a–b), which were defined by high fluorescence in the ssq1Δ mutants or in yeast treated with ethanol (Fig. 4c–h). Further co-localization assays were performed, using rhodamine 123 as an indicator of the mitochondrial membrane potential (Δψ). Images showed that in the WT strain, free Fe2+ fluorescence was observed in all cells, but with greatest intensity in mitochondrial structures, co-localized with O2− generation; interestingly, high-intensity fluorescence was observed inside the cytoplasmic membrane, a response that intensified under ethanol treatment (Fig. 4i–l). As expected, in ssq1Δ mutants, which showed affected respiration and Δψ behavior (see below), high free Fe2+ fluorescence was observed, but it was not co-localized with mitochondrial activity; however, it was probably associated with vacuolar structures (Fig. 4m–n). These results suggest that at least in the WT, the O2− generation and free Fe2+ release occurred in the mitochondria, although other cellular compartments may also have been involved in this effect. Co-localization assays were performed to analyze O2−/Fe2+ levels in ISC and iron-transport mutants. In both ssq1Δ, atx1Δ and aft1Δ mutants, increased levels of fluorescence corresponding to O2− generation and free Fe2+ release were observed; in yeast strains, these increased levels were co-localized in possible mitochondrial structures, around some hyper-structures that may correspond to vacuoles (Fig. 4o–t). These results confirm that ISC mutations cause an increase in free iron, which was enhanced by treatment with ROS inducers such as ethanol, and which was preferentially associated with mitochondrial structures; interestingly, in ISC and iron-transport mutants as well as in WT under ethanol treatment, a clear swelling of vacuolar structures was observed.

To confirm that enhanced ROS generation was correlated with the free Fe2+ content, real-time quantification of ROS by flow cytometry in a medium containing sufficient iron was performed using Fe2+ dose-response tests (Fig. 5). As expected, all yeast strains displayed a significant, dose-dependent increment in levels of fluorescence (indicating O2− and H2O2 generation) when treated...
with increased concentrations of Fe\(^{2+}\), ssq1Δ and isa1Δ mutants showed the highest ROS generation; in contrast, grx5Δ mutants showed a moderate increment in ROS compared to the WT strain, but this increment was lower than that observed in ssq1Δ and isa1Δ mutants. For atx1Δ, mrs4Δ, and aft1Δ iron-homeostasis deficient mutants, both the DHE and DHR123 probes showed that the ROS content was similar to those observed in the WT and grx5Δ strains (Fig. 5a–b). These results indicated that free iron is responsible for an important proportion of the ROS generated in both ISC and defective iron-transport mutants.

The role of free iron in ROS generation in the presence of ethanol in ISC mutants was further confirmed by determination of cell survival in cultures treated with a toxic concentration of ethanol (10%) and the addition of an iron chelator (10 \( \mu \)M phenanthroline). The ISC mutants showed decreased survival following ethanol treatment, and ssq1Δ and isa1Δ mutants were most sensitive; when a Fe\(^{2+}\) chelator was added to the cultures, survival was significantly restored in all strains (Fig. 5d). These results are in concordance with an increased free iron content and exacerbation of ROS generation observed in ISC mutants. Likewise, phenanthroline also exerted a significant protective effect against toxic ethanol concentrations in the iron homeostasis defective mutants atx1Δ, mrs4Δ, and aft1Δ. These results indicate that the toxic effects produced by ethanol and other oxidant agents were associated with intracellular Fe\(^{2+}\) release in a dose-dependent manner, and that this phenomenon is exacerbated by dysfunction of the ISC system.
Cellular free iron release is correlated with a decrement in Fe-S proteins content

The above results led us to hypothesize that the increment in free iron levels induced by oxidative stress and ethanol and exacerbated by ISC system dysfunction arise partially from iron sources such as proteins containing Fe–S centers, including complexes II and III of the ETC. Raman spectroscopy analysis has been used as an analytical tool for determination of Fe–S

Figure 4. Confocal microscopy images of S. cerevisiae cultures treated with oxidant agents to detect localization of ROS and free Fe$^{2+}$ in intracellular compartments. Yeast YPD-grown cultures were loaded with the fluorescent probe DHE or PGFL and treated with and without ethanol (10%), incubated for 30 min at 30 °C and co-loaded with Rho123 as a mitochondrial co-localization marker, and observed using a confocal microscope (Olympus FV1000). (A–D) WT and ssq1Δ mutant grown on glucose (2%) and with ethanol (10%) (E–H), using DHE probe for superoxide determination or PGFL probe for free Fe$^{2+}$ determination, as indicated. Cells are shown in boxes, and mitochondria and vacuoles are indicated with (m) and (v). (I–L) Images of WT yeast growth in glucose or treated with ethanol 10% and stained with PGFL and Rhodamine 123 probes, as indicated. (M–P) Images of ssq1Δ mutants treated with glucose or ethanol (10%), using PGFL and Rhodamine 123 probes, as indicated. (Q–T) Images of O$_2$ and free Fe$^{2+}$ co-localization in atx1Δ and aft1Δ mutants grown on glucose or treated with ethanol (10%), and staining with PGFL or DHE probes, as indicated. Superoxide generation areas are shown as fluorescent granules within the cells (see inset of A, J–L), free Fe$^{2+}$ is shown as green cells and green granules within the cells (see inset of B and I), merged images are shown as yellow cells and granules within cells (O–T), and mitochondrial structures are shown as cyan granules within the cells, using the Rho123 probe (I–L). Images of the yeast cells were taken using 10× magnification, 40× magnification, and 65× magnification of yeast cells.

doi:10.1371/journal.pone.0111585.g004
species and contents; therefore, this technique was utilized to
determine the Fe–S content in mitochondria isolated from
ISC mutants or the iron-transport defective mutants atx1D and
mrs4D grown in YPD. Signal intensities in the interval 200–700 cm
s1 at 632.8 nm in the Raman spectra are in agreement with signals
corresponding to photonic emission, characteristics of previously
described [2Fe–2S] and [4Fe–4S] centers [26,33]. In our system,
mitochondria from S. cerevisiae clearly showed Raman signals in
stretching regions of peaks at 345–365, 390–440, 460–480, 490–500,
510–520, and 640–660 cm
s1 (Fig. 6a). The intensity of
Raman signals of the Fe–S centers were clearly diminished in
mitochondria from ssq1D and isa1D mutants compared to WT
mitochondria, whereas in grx5A and mrs4A mutants, a decrease in
aconitase activity was also observed but to lesser extent that in ISC
mutants.

To investigate the role of Fe-S proteins in Fe
s2+ release under oxidative stress, we conducted a functional analysis of the 4Fe–4S
protein cis-aconitase, which has been described as ROS-sensitive
and as an iron donor for the Fenton reaction. With respect to
[2Fe–2S] clusters, we assayed the activity of ETC complex
III which is rich in that center and is known to be a main source of
O2
s2− generation in mitochondria [34]. Cis-aconitase activity was
almost totally abolished in the isa1D mutants compared to the WT
strain (Fig. 6b), confirming that in the ISC assembly system, the
Isa1 protein is essential for assembly of the [4Fe–4S] cluster into
aconitase enzyme. Moreover, aconitase activity was partially
inhibited in the remaining ISC mutants, whereas in mrs4A and
atx1D mutants, a decrease in aconitase activity was also observed
but to lesser extent that in ISC mutants.
However, respiratory complex assembly studies have indicated that the [2Fe–2S]-Rieske protein of complex III is essential for the correct formation of ETC supercomplexes, constituted of complex III (ubiquinol-cytochrome c reductase or $bc_1$ complex) and complex IV (cytochrome $c$ oxidase) [4,35]. Therefore, we analyzed the assembly of ETC supercomplexes using BN-PAGE gels [27–30]. Interestingly, the results of the BN-PAGE gels indicated that assembly of the $III_2IV_2$ and $III_2IV_1$ supercomplexes is dependent on the functionality of the ISC system. The band corresponding to the $III_2IV_2$ supercomplex was almost absent in mitochondria from $ssq1 \Delta$ and $isa1 \Delta$ mutants, but the band corresponding to the $III_2IV_1$ supercomplex was detected at low levels in all ISC mutants (Fig. 7a). Remarkably, in the densitometric analysis of the gels, the intensity of the bands corresponding to the $III_2IV_2$ supercomplex was significantly affected in $ssq1 \Delta$ and $isa1 \Delta$ mutants; the $III_2IV_1$ supercomplex was also diminished in $ssq1 \Delta$, $isa1 \Delta$, $grx5 \Delta$, and $mrs4 \Delta$ mutants, whereas in $atx1 \Delta$ mutant, a response similar to the WT was observed (Fig. 7b). In addition, densitometry data indicated that the content of dimeric complex V (i.e. the $F_1F_0$ ATPase) and its monomer was significantly increased in $grx5 \Delta$ mutants, whereas the dimer of complex IV and II remained unaffected in ISC mutants but not in the iron-transporter deficient strains. Immunoblotting assays using anti-Rip1 antibody (Rieske protein) confirmed the supercomplexes formation and Raman spectrometry findings. In western blot analysis of mitochondrial
Mitochondrial ETC functionality is affected by ISC mutations

Raman spectroscopy observations of mitochondria isolated from S. cerevisiae under iron sufficiency indicated that the [Fe–S] cluster content was diminished in ssq1Δ and isa1Δ mutants, and native gels showed a clear alteration in the amount of the III2IV2 and III2IV1 supercomplexes. We therefore evaluated the in situ mitochondrial oxygen consumption rates (OCR) to determine the functionality of the ETC in ISC mutants. In this case, atx1Δ and mrs4Δ mutants were included as controls, because atx1Δ cells showed decreased mitochondrial content and severely impaired function following isolation (data not shown). Respiration was completely abolished in both the coupled and uncoupled states in ssq1Δ and isa1Δ mutants, whereas in grx5Δ, atx1Δ, and mrs4Δ mutants, the OCR was partially decreased in comparison with the WT strain (Fig. 8a–b). Remarkably, ethanol treatment caused oxygen release in the assay chamber (i.e., negative values for OCR, Fig. 8d–f) instead of oxygen consumption in ISC mutants, except for grx5Δ and atx1Δ. This is suggestive of ROS production, since superoxide dismutase catalyzes the conversion of O2− to O2 and H2O2, while catalase converts the latter species into H2O and O2. In S. cerevisiae mitochondria, complex III is the only site of ROS generation in the ETC, since it lacks a rotenone-sensitive complex I, the other site of ROS production in the ETC of superior eukaryotes [35]. We exposed the cells to antimycin A, an inhibitor of complex III, to further explore the possible role of complex III in ROS generation. In the absence of ethanol, a small amount of O2 generation was detected in all ISC mutants except in grx5Δ, whereas in the presence of ethanol, higher rates of O2 generation were detected in these mutants, and O2 consumption was fully inhibited in grx5Δ cells (Fig. 8f). Conversely, antimycin A-insensitive oxygen consumption was observed in WT, grx5Δ, and atx1Δ cells in the absence of ethanol, and only in WT and atx1Δ cells in the presence of ethanol. In contrast to the behavior of ISC mutants, mrs4Δ mutant, which shows affected iron homeostasis, displayed an OCR similar to that of the WT strain in the presence of glucose in the coupled state (Fig. 8a). To further corroborate the degree of mitochondrial dysfunction in the various strains, we evaluated mitochondrial membrane potential (Δψ). In concordance with their inability to respire, ssq1Δ and isa1Δ mutants did not exhibit Δψ (Fig. 9a). Furthermore, OCR correlated with the magnitude of Δψ in WT, grx5Δ, atx1Δ, and mrs4Δ cells, since membrane potential was 2–4-fold higher in WT than in grx5Δ or atx1Δ and mrs4Δ mutants. The OCR in a coupled state followed a similar pattern (Fig. 8a and 9a), which in turn may be related to the decreased complex II signal observed for these mutants (Fig. 7a and 7b).

To determine which segment of the ETC was responsible for the effects described above, partial ETC reactions were analyzed. Succinate-DCIP oxidoreductase activity (representative of complex II activity) was observed to be abolished in ssq1Δ and isa1Δ mutants, and to be severely affected but not abolished in control mutants atx1Δ and mrs4Δ, which displayed behavior similar to grx5Δ mutants, that showed 30–40% of the WT activity (Fig. 9b).

The same trend was observed in both antimycin A-sensitive succinate-cytochrome c oxidoreductase (representative of the activity of complex III, using the endogenous ubiquinol-6 pool as a substrate) [24] and cytochrome c oxidase (representative of complex IV) activities; their activities in complexes III and IV were almost entirely abolished in ssq1Δ and isa1Δ mutants, whereas in
Antimycin A-sensitive succinate-cytochrome c oxidoreductase activity is dependent on electron transfer between complex II and complex III. Thus, to eliminate the possibility that impaired succinate-cytochrome c oxidoreductase activity was the result of impaired complex II activity, and not of direct damage of complex III, the activity of complex III was tested by reducing the mitochondrial quinone pool with glycerol via the concerted action of porin-associated glycerol kinase and mitochondrial glycerol-3-phosphate dehydrogenase [36,37]. No differences in this activity with respect to mitochondria from WT cells were observed in grx5Δ, atx1Δ, or mrs4Δ mutants, while mitochondria from ssq1Δ and mrs4Δ mutants exhibited a 3-fold diminution (Fig. 9d).

Another respiratory enzyme that utilizes cytochrome c as its electron acceptor is L-lactate-cytochrome c oxidoreductase or cytochrome b2 [38]. This activity was similar in all the mutants tested; however, it was three-fold lower in the mutants than in the WT strain (Fig. 9e).

Discussion

The toxic effects of ethanol on mitochondrial function have been attributed to a variety of factors, ranging from alteration of ETC complex activities [39], loss of heme groups from cytochromes [40], oxidative degradation and depletion of mtDNA [41], diminished number of active ribosomes [34], decreased glutathione pools and lipid peroxidation [42]. Although many of these events are related to increased ROS generation, none of these reports have addressed the possibility that disturbances in mitochondrial free iron from Fe–S clusters may be related to ethanol toxicity. Nevertheless, earlier reports have identified a link between ROS overproduction and iron overload in the toxic effects of ethanol, since iron chelation attenuates some of the disturbances in the antioxidant defenses caused by ethanol consumption [43,44]. We have recently observed in yeast cells that ROS generation during ethanol stress was exacerbated by mutations in ISC, which participate in the various steps in Fe–S biogenesis [20]. In the present study, we hypothesized that the increased free iron content was correlated with the dysfunctional ISC assembly system, which affects iron homeostasis after treatment with ethanol or ROS generators. The results obtained in the present study indicate that a dysfunctional ISC assembly system increases susceptibility to ethanol and ROS generators. The results obtained in the present study indicate that a dysfunctional ISC assembly system increases susceptibility to ethanol and ROS generators, such as H2O2 and menadione (Fig. 1), probably via both increased mitochondrial ROS levels and disruption of ETC functionality. These effects seem to be linked to an increment in the free iron pool, since iron chelation had a protective effect in ISC mutants exposed to ROS inducers. This suggests a close relationship between levels of free iron and ROS generation, where the Fe–S assembly system plays a relevant role in iron homeostasis and its dysfunction may partially contribute to excessive iron release to the cell.

The sensitivity of the mutants to ROS generators and ethanol, as well as their levels of ROS production and free iron release (Fig. 1–3), appear to be highly dependent on the functionality of the ISC assembly system.
Role of ISC System on Iron-Dependent ROS Generation and ETC Function

Figure 9. Analyses of the functionality of mitochondrial respiratory chain complexes in S. cerevisiae ISC mutants. Mitochondrial functionality was evaluated in mitochondrial suspensions obtained from cultures grown in liquid YPD medium, cells were harvested in the late exponential growth phase, mitochondria were isolated and re-suspended in the appropriate buffer, and mitochondrial activities were measured as described in the Materials and Methods. A) membrane potential, B) activity of succinate-DCIP oxidoreductase, C) activity of succinate-cytochrome c oxidoreductase, D) activity of glycerol-cytochrome c oxidoreductase, E) activity of L-lactate-cytochrome c oxidoreductase, and F) activity of cytochrome c oxidase. Values are the mean of three independent experiments. SE values are indicated as bars (n = 3), one-way ANOVA with Tukey’s post-hoc test was used to compare yeast strains, and significant differences (p<0.05) are indicated with different lowercase letters. doi:10.1371/journal.pone.0111585.g009

The findings obtained using the fluorescent ROS probes DHE and DHR123 (Fig. 2) suggest that ROS are mainly generated and probably accumulated in mitochondria after treatment with toxic concentrations of ROS inducers; an additive effect on ROS generation was observed in ISC mutants. In addition, the results obtained for act1Δ, mrv4Δ, and act1Δ mutants confirmed that altered iron homeostasis caused an increased generation of ROS (such as superoxide and hydrogen peroxide), which was exacerbated when ethanol, H2O2, or menadione were used as inducers, suggesting the participation of free iron in the ROS sensitivity (Fig. 2).

Biogenesis of Fe–S centers in S. cerevisiae occurs mainly in mitochondria, and the assembly mechanism of these centers depends on the functionality of ISC gene products. Iron is an essential component of this process, and its cellular content is dependent on transport systems, chelating proteins and storage. It is well known that iron can be released from Fe–S proteins by O2•− or H2O2 [1,5,12,29], which in turn can lead to the generation of the strongly oxidant OH• radical via Fenton’s chemistry.

The findings for Fe2+ release in ssq1Δ and isa1Δ mutants suggest that accumulation of preassembled Fe–S clusters, due to insertion failure caused by disruption of ISC or the absence of the target apoproteins (i.e., mitochondrial Fe–S-containing proteins), leads to an increment in the free Fe2+ pool, provoking an oxidative stress event (Fig. 3). This may lead to denaturing/dissociation of already assembled Fe–S hemoproteins as an iron source; the prosthetic groups of these proteins could also be one of the iron sources that cause the iron imbalance, leading to a vicious circle of ROS generation (Fig. 10). This is concordant with the impairment of cytochrome b2, a heme enzyme, observed in all mutants tested. Additionally, it has been reported that in yeast mitochondrial superoxide dismutase (SOD2) can be metallated with iron instead of manganese when iron homeostasis is disrupted, leading to enzyme inactivation. Thus, SOD2 inactivation derived from the intramitochondrial free iron increment may be an additional factor involved in superoxide accumulation, since this metallation is more apparent in Grx5p and Ssq1p mutants, and affects downstream steps in iron-sulfur biogenesis [45,46].

The fact that excess iron in ssq1Δ and isa1Δ mutants causes an additive increase in ROS generation (Fig. 3) is in agreement with the increment in O2•− generation observed in these mutants in the presence of glucose plus 10% ethanol [Fig. 8f]. Because the release of Fe2+ was enhanced in these mutants, OH• radicals may have also been formed due to the Haber–Weiss cycle. This event would lead to a worsening of the redox state with catastrophic consequences for the cell, due to the high reactivity of OH• with virtually any class of biomolecule. In summary, we suggest that the iron-mediated mechanism of ROS inducer toxicity in ssq1Δ and isa1Δ mutants may be also be the result of impaired electron transfer at complexes II and IV (Fig. 9b and 9f), which in turn leads to a reduction in electron transporters in complex III, as
reflected by the effect of antimycin A in the Fig. 8f, and further generation of ROS. The increment in ROS may enable the release of more iron from storage systems, and probably from the prosthetic groups of Fe–S or heme proteins, contributing in this way to a feedback mechanism for ROS generation via the increment in the free iron pool (Fig. 10). Moreover, this may be related to an apoptotic phenotype in \textit{ssq1}\textsuperscript{D} and \textit{isa1}\textsuperscript{D} mutants, observed in the presence of toxic quantities of the ROS inducer, ethanol [20]. These suggestions are supported by the fact that grx5\textsuperscript{D}, in which lower levels of Fe\textsuperscript{2+} were released in the presence of ethanol, still displayed OCR and partial activity of all ETC complexes, as well as null oxygen release even in the presence of antimycin A (Figs. 8 and 9). The effects of Grx3p deletion suggest that the main function of this protein is the transitory storage of Fe–S clusters during ISC assembly, which is reflected by the lower iron release and resultant lower ROS generation observed in the grx5\textsuperscript{A} mutant.

Importantly, \textit{isa1}\textsuperscript{A} and \textit{ssq1}\textsuperscript{A} mutants showed similar phenotypes for ROS susceptibility, ROS generation, Fe\textsuperscript{2+} release, OCR, and ETC complexes activities. We speculate that the free iron released may originate from Fe–S-containing proteins such as ETC complex II, because the \textit{isa1}\textsuperscript{A} and \textit{ssq1}\textsuperscript{A} mutants were the major producers of ROS, and both completely lack complex II activity (Fig. 9b), suggesting that Fe–S-dependent recipient proteins are involved. These facts also suggest that the Isa1 protein could be involved in \textit{de novo} Fe–S assembly or recycling of Fe–S-containing proteins from complex II, because Isa1p/Isa2p have also been described as iron reservoirs [12]. However, it is possible that the iron release mainly proceeds from iron storage sources, such as vacuoles or iron-chelation proteins, such as frataxin, which make iron already present within mitochondria available for Fe–S oxidation.

\textbf{Figure 10. Model proposed for the mechanism of mitochondrial generation of ROS dependent on free Fe\textsuperscript{2+} release from Fe–S-containing proteins in \textit{S. cerevisiae}}. The [2Fe–25] commonly carried by the multi-protein complex (Ssq1–Jac1–Mge1–Grx5) can also be assembled into recipient apoproteins, such as the Rieske protein of cytochrome bc\textsubscript{1} from respiratory complex III. The Isa1 and Iba57 proteins may function as iron reservoirs, from which the metal can subsequently be transferred to [Fe–S] centers or heme prosthetic groups from cytochrome bc\textsubscript{1}. When superoxide (O\textsubscript{2}–) is generated by electron leaking in the ETC, and other ROS are produced by oxidative metabolism or by oxidant agents, the [4Fe–4S], [3Fe–4S], or [2Fe–25] clusters contained in the ETC complexes are disrupted. This event provokes a generalized uncoupling/denaturation of Fe–S proteins, causing a release and thus an increment in the iron labile pool (Fe\textsuperscript{2+}), which increases mitochondrial ROS levels via the Fenton and Haber–Weiss reactions. If the ISC assembly system is dysfunctional, supercomplex (III\textsubscript{IV}) formation is affected, as is [Fe–S] recycling, provoking ETC dysfunction. Hence, the levels of ROS generation increase in an additive manner by a vicious circle of disruption of iron-containing or iron storage proteins, causing an imbalanced ROS content (increment of species such as H\textsubscript{2}O\textsubscript{2} and superoxide species), that provokes mitochondrial dysfunction and may ultimately lead to apoptotic events.

doi:10.1371/journal.pone.0111585.g010
cluster synthesis when iron concentrations are low or the Fe–S content is diminished [16].

With the objective to elucidating this last hypothesis, concerning the roles of Isa1 in Fe–S assembly and iron recycling, three types of possible recipient proteins were evaluated: (1) the [4Fe–4S] cluster from cys-aconitase, which has been described as highly sensitive to ROS and which is considered to be an iron donor for the Fenton reaction; (2) the Rieske protein from ETC complex III, which is rich in [2Fe–2S] clusters, and which is recognized as the main source of superoxide generation in mitochondria [47], (3) along with succinate dehydrogenase from complex II, which contains [2Fe–2S], [3Fe–4S], and [4Fe–4S] clusters [48]. As shown in Fig. 6a, cys-aconitase was not affected in atx1Δ or mrs4Δ mutant, but it was affected in Isa1 mutants (sq1Δ and grx5Δ), and it was almost totally abolished in the isa1Δ strain. Concordant with our results, a reduction in aconitase protein expression and enzymatic activity has been described in mrs4Δ iron-transport mutant, along with iron dependence for de novo Fe–S cluster formation, dependent on the Mrx3/Mrs4 iron-transporters or frataxin involved in iron homeostasis (its diverse functions include Fe–S clusters synthesis, heme biosynthesis, aconitase repair, respiratory regulation, iron detoxification, iron storage, and oxidative stress protection) [15–17]. These findings confirm that in [4Fe–4S] clusters, assembly into target proteins such as cys-aconitase is dependent on the Isa1 protein, as described previously [12]. Interestingly, grx5Δ mutant showed a diminution in complex II activity, suggesting that assembly of the Fe–S clusters of complex II may also be assisted by the Grx5 protein. Abolishment of succinate dehydrogenase activity (null activity of complex II) in the isa1Δ strain strongly suggests its involvement in assembly of Fe–S clusters in the Sdh protein of complex II (Fig. 7a). Interestingly, our Raman spectrophotometry results were in agreement with these observations, indicating that the [2Fe–2S] and [4Fe–4S] cluster content in isolated mitochondria was severely diminished in isa1Δ and sq1Δ mutants, compared with the WT. Meanwhile, in the conditional iron-transport mutants atx1Δ and mrs4Δ, and in grx5Δ strain mutants, an increased content of Fe–S clusters was observed (Fig. 6a). Iron-deficiency has been described to provoke a decrease in the iron uptake in atx1Δ mutants [31], a condition that causes an increment in iron uptake, as in mrs4Δ mutants [17]. This iron-limiting condition induces upregulation of iron-dependent genes controlled by the transcriptional regulator Atf1 [19], and in some genes involved in the ISC system [50]; these observations are concordant with the increased [2Fe–2S] and [4Fe–4S] clusters, corresponding to signals in the Raman spectra of mitochondria from atx1Δ, mrs4Δ, and grx5Δ mutants.

The assembly of ETC supercomplexes formed by cytochrome bc1 complex and cytochrome c oxidase (complexes III and IV of the ETC) is dependent on the integration of Rieske protein into the bc1 complex [4,35,51]. In this context, we examined formation of ETC supercomplexes III/IV in S. cerevisiae ISC mutants using native-blue gels [27–30]. We found that supercomplex IIIcIVc2 was virtually undetectable in mitochondria isolated from sq1Δ and isa1Δ mutants, but supercomplex IIIcIVc1 was diminished in both ISC mutants, as well as in grx5Δ mutants (Fig. 7). In agreement, with these findings, immunoblotting using an anti-Rip1 antibody showed that the ISC system is involved in the assembly of the Rieske subunit of complex III, and that Ssq1p is essential, but Isa1p and Grx5p are not. Moreover, in grx5Δ, atx1Δ, and mrs4Δ mutants, the Rieske protein was up-regulated (Fig. 7c). This result is coincident with the increase in Fe–S cluster signals observed in the Raman spectra (Fig. 6). These findings suggest that the ISC system plays an important role in the assembly of supercomplexes III/IV of the ETC, probably in a Rieske-dependent manner. Alternatively, these findings suggest that these proteins are related to heme biogenesis or assembly in ETC cytochrome-containing proteins such as complex IV or cytochrome b2, by modulating the bioavailability/recycling of iron. However, further studies are required to corroborate these hypotheses in detail.

We expected that disruption of ISC biogenesis would only impair complexes containing Fe–S clusters. This expectation was corroborated with the observed abolishment of complex II activity in sq1Δ and isa1Δ mutants, as this complex contains [2Fe–2S], [3Fe–4S], and [4Fe–4S] clusters in the catalytic dimer of the enzyme that play a central role in catalysis, as they receive electrons from FADH2 and transfer them to the membrane domain where quinone reduction occurs [24]. Interestingly, the activities of complexes II and III in grx5Δ mutants, although decreased, were not totally abolished, as in the other ISC mutants; indeed, the activity of complex III with glycerol was similar to that of the WT strain (Fig. 9d). This result seems to contradict the essential role of Grx5p in the activity of complex II, reported by Rodríguez-Manzaneque et al. (2002) [9]. However, this discrepancy may be attributed to the different technique used to measure complex II activity in that study, which involved following the formation of formazan resulting from the reduction of a tetrazolium salt by complex II [52], formazan can also be formed by the reduction of tetrazolium salts by ROS [53]. We avoided this issue by monitoring reduction of DCIP. It has been proposed that Grx5p participates in ISC cluster biogenesis by assisting in the transference of Fe–S clusters from the scaffold to target proteins [54] and/or by repairing mixed disulfides between glutathione and ISC assembly factors [55], in addition, it may constitute transitory Fe–S cluster storage [56], although its exact role remains to be elucidated. Thus, it appears that the function of Grx5p in the assembly of Fe–S clusters from complexes II and III may partially be replaced by other glutaredoxins.

Collectively, these results indicate that the respiratory incompetence of sq1Δ and isa1Δ mutants was mainly due to defective oxidation of substrates in complex II and impaired delivery of electrons to complex III, and to oxygen in complex IV.

Although they lack Fe–S clusters, both complex IV and cytochrome b2 (i.e. both heme-containing proteins) were also affected in all ISC mutants, except grx5Δ (Fig. 9). This was not at all surprising, since in yeast decreased heme biosynthesis and cytochrome deficiency are general phenotypic features of cells with an impaired mitochondrial ISC system [6]. ISA1 and SSQ1 yeast mutants contain decreased amounts of both hemes c+c1 and b and residual activity of cytochrome c oxidase [57]. Concerning the null complex IV activity of sq1 and isa1 mitochondria, a similar phenotype was reported by Gelling et al. (2008) [10]. The impaired complex IV activity observed in these mutants may also be explained by the inability of this enzyme to form supercomplexes with complex III, as it has been demonstrated that when complex III has an incorrect conformation, the activity of complex IV may be strongly affected [4,35,51]. This is in full agreement with the impaired formation of supercomplexes IIIIV2 and IIIIV1 in ISC mutants (Fig. 7).

Mitochondrial respiration was absent in sq1Δ and isa1Δ mutants in all respiratory states. This effect seems to be mainly attributable to defective electron transfer to O2 at complex IV, since this activity was fully inhibited in these cells (Fig. 8). Another factor contributing to the respiratory incompetence of these mutants is their inability to oxidize substrates at complex II, as demonstrated by their observed null succinate-DCIP oxidoreductase and succinate-cytochrome c oxidoreductase activities. The impairment in the latter activity was not attributed to defective
electron transfer at complex III, since we detected cytochrome c reduction in isa1Δ in the presence of glycerol at the same level as in WT, but at significantly diminished levels in sqs1Δ mutants (Fig. 8). This is concordant with the fact that Isa1p has not been described as participating in the assembly of 2Fe–2S centers, which is the type of ISC present in the Rieske subunit of complex III [5]. However, Raman spectra indicated that a decrease in Fe–S protein content in mitochondria occurs in sqs1Δ and isa1Δ mutants, also associated with loss or diminution of the Rieske protein in complex III, although the involvement of mitochondrial proteins containing heme groups cannot be neglected. The last observation is in accordance with formation of supercomplexes in the ETC that depend on Fe–S proteins, as seen in the mitochondrial extract from sqs1Δ and isa1Δ mutants in native gels and western blots, suggesting that, in addition to Fe–S cluster assembly, these proteins could also be involved in heme assembly in mitochondrial respiratory complexes.

Regarding the role of the functionality of ETC in ethanol tolerance, it must be considered that in yeast, mitochondria participate in the maintenance of the redox balance during metabolism of sugars by oxidizing the NADH generated during both glycolysis and ethanol oxidation by the cytosolic and mitochondrial isoforms of alcohol dehydrogenase [58]. This process is very important to avoid deleterious production of mitochondrial ROS, since high NADH/NAD+ ratios favor higher rates of ROS production in the ETC [59]. In yeast, the NADH dehydrogenases Nde1, Nde2, and Ndh1 shuttle electrons from NADH to the quinone pool [60], and the ubiquinol generated is oxidized by the quinol-oxidase site of complex III. Although sqs1Δ and isa1Δ mutants exhibited partial or full complex III activity (Fig. 8), it is possible that the null activities of complexes II and IV indirectly interfere with NADH oxidation and ethanol metabolism. This prevents the re-oxidation of electron acceptors in complex III, which in turn may lead to an increase in the generation of semiquinone radicals, favoring the generation of O2•−. In accordance with this notion, it has been demonstrated that the inhibition of complex II or complex IV may enhance mitochondrial ROS generation [61,62]. This point is further supported by the fact that the addition of antimycin A leads to ROS generation (Fig. 8c and 8f). This also indicates that ubiquinol is being oxidized at the quinol oxidase (Qo) site of complex III for bifurcated reduction of cytochrome b and posterior O2•− formation by inhibition of re-oxidation of cytochrome b562 in the quinone reductase site (Qr) induced by antimycin A [24]. Otherwise, the generation of O2•− by antimycin A would not be possible. Furthermore, the exacerbation of O2•− release by antimycin A with ethanol treatment (Fig. 8) is suggestive of further impairment of the Q cycle in complex III, corroborating the hypothesis that the toxicity of ROS inducers is mediated by altered electron transfer in the ETC, leading to enhanced ROS generation. It must be stressed again that O2•− generation is an indicator of ROS production because O2•− is a product of the degradation of O2•− and H2O2, catalyzed by superoxide dismutase and catalase, respectively. Importantly, we also found that these mutants have increased catalase activity [20], which may be an adaptive response to enhanced ROS generation due to impaired ETC function.

In summary, exacerbation of ROS generation in S. cerevisiae caused by treatment with stressors such as ethanol, H2O2, and menadione, occurs via Fe2+/2− release, which is favored by an iron-dependent ROS generation cycle. Microscopic analysis of the WT strain showed that free Fe2+/2− release and ROS co-localized mainly in mitochondria, and were exacerbated by ethanol treatment (a ROS inducer), whereas in sqs1Δ mutants, both free Fe2+/2− and ROS were observed in all cells. This pattern was also observed in atx1Δ and mrs4Δ mutants, which are hyper-iron accumulators in an iron-rich media. Interestingly, a phenotype of bloated vacuole structures was observed in ISC mutants, as in iron-accumulator mutants (atx1Δ and mrs4Δ), suggesting dysfunctional iron homeostasis associated with mitochondria and vacuole organelles.

Raman spectroscopy and supercomplex formation of mitochondria isolated from ISC mutants indicated that disruption of Sqs1p and Isa1 proteins provoked a decrease in [2Fe–2S] and [4Fe–4S] cluster content that was reflected in loss of the Rieske protein from complex III and disrupt supercomplex formation between complexes III and IV, leading to dysfunction of the ETC and probably to mitochondrial apoptotic events.

Our findings indicate that free Fe2+/2− release and ROS generation are interdependent and are associated with mitochondrial iron homeostasis, via Fe–S-containing proteins, and with storage/detoxification systems, such as frataxin, which are important iron sources. The bloated vacuoles observed in ISC mutants following treatment with ROS inducers, as well as in iron-accumulator mutants (atx1Δ and mrs4Δ) suggest that an iron imbalance occurred in the loss of iron homeostasis, that in turn contributed to ROS generation, and to impaired Fe–S cluster biogenesis of proteins from the ETC. The oxidative stress generated and the effects on the Fe–S-containing proteins led to mitochondrial dysfunction.

Supporting Information

Figure S1 Studies of growth in plates of the S. cerevisiae ISC mutants. A–D) Dilutions of yeast suspensions were cultured on YPD agar plates with or without ROS-inducers at the indicated concentrations at 30°C for 48 h. Yeast cultures grown on YPD medium plates with: A) different concentrations of the iron chelator 1,10-phenanthroline (20 μM) and ethanol (8%). B) low-iron content using the iron chelator 1,10-phenanthroline (20 μM) and different concentrations of ferrous iron (5–20 μM). C) ROS-inducers at indicated concentrations of H2O2 (4 mM), menadione (80 μM), and ethanol (3%). D) low-iron content using 1,10-phenanthroline (20 μM) plus ferrous iron (20 μM) with the concentrations indicated of ROS-inducers H2O2 (4 mM), menadione (80 μM), and ethanol (3%). E) low-iron content using 1,10-phenanthroline (20 μM) plus ferrous iron (20 μM) with the concentrations indicated of ROS-inducers H2O2 (4 mM), menadione (80 μM), and ethanol (3%).

(TIF)

Acknowledgments

We thanks to Drs. G. Del Rio and S. Funes from Instituto de Fisiología Celular/UNAM by yeast strains donation. To Conte L. and Zara V. by Anti-Rip1 antibody donation.

Author Contributions

Conceived and designed the experiments: RVPG CCR JCG. Performed the experiments: RVPG MG LAS ALDP. Analyzed the data: RVPG MVM CCR JLR SJS FR JSRZ JCG. Contributed reagents/materials/analysis tools: JLR SJS FR ASM JSRZ. Wrote the paper: CCR JCG.
References

1. Schilde B, Voinche C, Beinert H, Craig E (1999) Evidence for a conserved system for iron metabolism in the mitochondria of Saccharomyces cerevisiae. Proc Natl Acad Sci USA 96:10206–10211.

2. Hertz-Köhl KG, Sítiková JF, Vojtěch LE (2006) Interaction of the iron-sulfur cluster assembly protein IscU with the Hsc66/Hsc20 molecular chaperone system of Escherichia coli. Proc Natl Acad Sci USA 99:7790–7795.

3. Dunkewicz R, Schilde B, Kneszier H, Walter W, Craig EA, et al. (2003) Ssig, a mitochondrial Hsp70 involved in iron-sulfur (Fe/S) center biogenesis. Similarities to and differences from its bacterial counterpart. J Biol Chem 278: 29719–29727.

4. Conte L, Sara V (2011) The Riceye Iron-Sulfur Protein: Import and Assembly into the Cytochrome bc1 Complex of Yeast Mitochondria. Biologery 36:3941.

5. Lill R, Hoffmann U, Molik S, Perik AJ, Rätzschel N, et al. (2012) The role of mitochondrial iron in cellular iron-sulfur protein biogenesis and iron metabolism. J Biol Chem 287:1491–1500.

6. Lill R, Mühlenhoff U (2006) Iron-sulfur protein biogenesis in eukaryotes: components and mechanisms. Annu Rev Cell Dev Biol 22: 457–496.

7. Lill R (2009) Function and biogenesis of iron-sulfur proteins. Nature 460:183–191.

8. Schilde B, Williams B, Kneszier H, Pukuta S, D’Silva P, et al. (2006) Evolution of mitochondrial chaperones utilized in Fe-S cluster formation. J Biol Chem 16:1060–1065.

9. Rodriguez-Manzoque MT, Tamari J, Belli G, Ros J, Herrero E, etc. (2002) Grox5 is a mitochondrial glutaredoxin required for the activity of iron/sulfur enzymes. Mol Biol Cell 13:1109–1121.

10. Gelling C, Davies IW, Richardaert L, Lill R, Mühlenhoff U (2008) Mitochondrial Biogenesis is required for Fe/S-cluster formation on aconitase and activation of radical SAM enzymes. Mol Cell 28:1831–1841.

11. Sheitel AD, Wilbrecht N, Weillier, B, Blaesser EP, et al. (2012) The human mitochondrial ISCA1, ISCA2, and ISB7 proteins are required for Fe/S protein maturation. Mol Biol Cell 23:1157–1166.

12. Mühlenhoff U, Richter N, Pines O, Pierik AJ, Lill R (2011) Specialized function of yeast Isa1 and Isa2 proteins in the maturation of mitochondrial [4Fe-4S] clusters. J Biol Chem 286:41205–41216.

13. Turrens JF (2003) Mitochondrial formation of reactive oxygen species. J Physiol 56:335–344.

14. Hoffmann B, Uzarska MA, Berndt C, Godoy JR, Haunhorst P, et al. (2011) The mitochondrial iron-sulfur cluster assembly system. Complexes of Yeast Respiratory Metabolism in M. Lill and U. Mühlenhoff (eds), Springer, New York.

15. Turrens JF (2003) Mitochondrial formation of reactive oxygen species. J Physiol 56:335–344.

16. Zhang Y, Lyver ER, Knight SA, Pain D, Lesuisse E, et al. (2006) Mrs3p, Mrs4p, components and mechanisms. Annu Rev Cell Dev Biol 22: 457–486.

17. Zhang Y, Lyver ER, Knight SA, Pain D, Lesuisse E, et al. (2006) Mrs3p, Mrs4p, components and mechanisms. Annu Rev Cell Dev Biol 22: 457–486.

18. Muñoz-Pinedo C, Escobar-Del-Rio R, Bleda JL, and de Pablos L (2007) Mitochondrial iron transport and metabolism. Annu Rev Nutr 27:279–292.

19. Escobar-Del-Rio R, Bleda JL, and de Pablos L (2007) Mitochondrial iron transport and metabolism. Annu Rev Nutr 27:279–292.
57. Lange H, Mühlenhoff U, Denzel M, Kispal G, Lill R (2004) The heme synthesis defect of mutants impaired in mitochondrial iron-sulfur protein biogenesis is caused by reversible inhibition of ferrochelatase. J Biol Chem 279:29101–29108.

58. Bakker BM, Bro C, Kotter P, Luttik MA, van Dijken JP, et al. (2000) The mitochondrial alcohol dehydrogenase Adh3p is involved in a redox shuttle in Saccharomyces cerevisiae. J Bacteriol 182:4730–4737.

59. Murphy MP (2009) How mitochondria produce reactive oxygen species. Biochem J 417:1–13.

60. Melo AM, Bandeiras TM, Teixeira M (2004) New insights into type II NAD(P)H:quinone oxidoreductases. Microbiol Mol Biol Rev 68:603–616.

61. Ferguson M, Mockett RJ, Shen Y, Orr WC, Sohal RS (2005) Age-associated decline in mitochondrial respiration and electron transport in Drosophila melanogaster. Biochem J 390:501–511.

62. Drose S (2013) Differential effects of complex II on mitochondrial ROS production and their relation to cardioprotective pre- and postconditioning. Biochim Biophys Acta 1827:578–587.