Characterization of phloroglucinol derivatives and diterpenes in *Euphorbia ebracteolata* Hayata by utilizing ultra-performance liquid chromatography/quadrupole time-of-flight mass spectrometry

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**Abstract** *Euphorbia ebracteolata* Hayata (*E. ebracteolata*) is a Chinese herbal medicine used for the treatment of tumor diseases. An ultra-performance liquid chromatography/quadrupole time-of-flight mass spectrometry (UPLC–Q–TOF/MS) based chemical profiling approach was established for the rapid separation and characterization on phloroglucinol derivatives and diterpenes in *E. ebracteolata*. Three phloroglucinol derivatives and nine diterpenes were identified by exact mass measurement and were further confirmed by Ms² data. In addition, the chemical profiles of six compounds were acquired by reference standards. Furthermore, the fragmentation rules of phloroglucinol derivatives and diterpenes of *E. ebracteolata* were analyzed, and each chromatographic peak was classified.

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1. **Introduction**

*Euphorbia ebracteolata* Hayata, a Chinese herb Langdu, has an effect of dissipating binds and insecticidal action. It is mainly used for treatment of scrofula, psoriasis [1]. The major constituents of *E. ebracteolata* include terpenoids, phloroglucinol derivatives, flavonoids, and sterols [2]. In the present study, we focus on the phloroglucinol derivatives and diterpenes. Diterpenes play an important role in the prevention of human diseases. The most
prominent effects of the diterpenes in *E. ebracteolata* are antitumor, antibacteria and immune adjustment [3–5]. The structures of these phloroglucinol derivatives and diterpenes are summarized in Scheme 1. Conventional phytochemical approaches, which are time-consuming, have been used for studying the structures of phloroglucinol derivatives and diterpenes in *E. ebracteolata* [6–9]. To our knowledge, ultra-performance liquid chromatography/quadrupole time-of-flight mass spectrometry (UPLC–Q–TOF/MS), as a powerful hypenated technique, has never been used to elucidate the structures of active compounds in *E. ebracteolata*.

In this study, we reported on the identification of phloroglucinol derivatives and diterpenes in the crude extract of *E. ebracteolata* by UPLC–Q–TOF/MS in positive-ion mode. Six reference chemicals, for example 2,4-dihydroxy-6-methoxy-3-methyl-acetophenone, were selected for characterization of the main components of *E. ebracteolata* for the first time. The mass spectral fragmentations of phloroglucinol derivatives and Jolkinolides abietane lactone in *E. ebracteolata* were discussed as well.

2. Materials and methods

2.1. Chemicals and materials

HPLC grade acetonitrile (Merck, Darmstadt, Germany), formic acid (CNW, Germany) were used, and ultra-pure water was prepared using a Milli-Q water purification system (Bedford, MA, USA). Ethanol (AR grade) for sample extraction was purchased from Guangzhou Chemical Reagent Factory (Guangzhou, China). All solvents were filtered through 0.22 µm membrane filters before analysis.

![Scheme 1](image)

Scheme 1  A proposed fragmentation pathway of Jolkinolide B.

![Fig. 1](image)

Fig. 1  HPLC and ESI-MS chromatograms of *E. ebracteolata*: (A) HPLC chromatogram at 226 nm and (B) ESI-MS chromatogram in the positive mode.
The herbal medicine *E. ebracteolata* was obtained from Liaoning Institute for Food and Drug Control (China) and was identified by chief pharmacist Boying Liu. A voucher specimen was deposited in Guangzhou Institute for Drug Control (Guangzhou, China). The reference standards of 2,4-dihydroxy-6-methoxy-3-methylacetophenone, 2,2′,4,4′-tetrahydroxy-6,6′-dimethoxy-3,3′-dimethyl-7,5′-bisacetophenone, 3,3′-diacetyl-4,4′-dimethoxy-2,2′,6,6′-tetrahydroxydiphenylmethane, Jolkinolide A, Jolkinolide B and Yuexiandujisu D were obtained from Institute for Chinese Medicine and Natural Medicine, Pharmacy College, Jinan University (Guangzhou, China).

2.2. Sample preparation

The herbal materials were ground into a fine powder (through a sieve of 100 meshes). An aliquot of 1 g of *E. ebracteolata*, accurately weighed, was extracted for 30 min with 25 mL of ethanol on an ultrasonic water bath (30 °C), and the extract was then filtered through a 0.22 μm syringe filter. The successive filtrate was used as the test solution.

2.3. Analytical method

Liquid chromatography was performed with an Agilent-1200 UPLC system (Agilent Corp., USA), equipped with a quaternary pump, UV detector, autosampler and column compartment. The column was an Agilent ZORBAX SB-C18 column (3.0 mm × 100 mm, I.D., 5 μm). The mobile phase consisted of (A) 0.1% formic acid in water and (B) acetonitrile (ACN). The UPLC elution condition was optimized as follows: 30%–65% B (0–20 min), 65%–75% B (20–30 min), 75%–85% B (30–35 min), 85% B (35–60 min) and isocratic at 30% B (60–63 min). The flow rate was at 0.4 mL/min. The column was maintained at 20 °C, and the injection volume of reference compounds and samples was 2 μL. The analytes were monitored at 226 nm.

LC–MS detection was performed directly after UV measurements, which was performed by an Agilent 6538 UHD Accurate-Mass Q–TOF/MS (Agilent Corp., USA) equipped with an ESI interface. The TOF/MS analysis was performed using full scan mode and mass range was set at m/z 50–1700 in positive-ion mode. The conditions of ESI source were as follow: drying gas (N₂) flow rate, 10 L/min; drying gas temperature, 350 °C; Nebulizer, 40 psig, capillary voltage, 4000 V; fragmentor voltage, 175 V; skimmer voltage, 65 V; octopole RF, 750 V. All the data were processed by Agilent MassHunter Software Ver. B. 03.01. Tuning mix was used for lock mass calibration in our assay.

3. Results and discussion

3.1. Optimization of HPLC and mass spectrometry conditions

To obtain desirable HPLC chromatograms, the procedure of sample preparation was optimized in selecting the extraction solvent. Three concentrations of aqueous ethanol, 50% (v/v in water), 70% and 95% were used for sample preparation. As no significant difference was observed among the obtained chromatograms, we selected 95% aqueous ethanol as the extraction solvent for this study. Among the five columns (Zorbax SB-C18, Dikma-C18, Ecosil-C18, Grace-C18 and Phenomenex-C18) that were tested for the separation of the sample, Zorbax SB-C18 column gave the best chromatographic resolution. 0.03% (v/v)
formic acid was added to the mobile phase to improve the mass spectrometry ionization efficiency and enable symmetric peak shapes. The UV detection wavelength was set at 226 nm, at which most components can be detected sensitively. Both the positive and negative ion modes were tested for MS analysis. Most phloroglucinol derivatives and diterpenes showed much cleaner mass spectral background and higher sensitivity in the positive mode than in the negative mode. The HPLC chromatogram and LC/MS total ion current of *E. ebracteolata* extract are given in Fig. 1.

Fig. 2 Chemical structures of identified compounds.
Fig. 3  The MS² spectra of compounds 1, 8 and 10.

Fig. 4  The MS² spectra of compounds 2 and 3.
3.2. Rationale for the characterization of phloroglucinol derivatives and diterpenes

The identity of known compounds in the herbal extract was confirmed by comparing with reference standards according to the retention time and MS² spectra [15]. The unknown compounds were characterized by analyzing their fragmentation behaviors in MS² spectra and by referring to the available literature information. A total of 12 compounds were characterized (Table 1), six (1, 2, 8, 9, 10, and 11) of which were identified by comparing the mass spectra and retention times with those of reference standards. Jolkinolide E was reported from E. ebracteolata for the first time. The structures of confirmed compounds are shown in Fig. 2.

3.3. Identification of phloroglucinol derivatives (compounds 1, 8 and 10)

Phloroglucinol derivatives were acetyl phenols containing hydroxyl, methoxy and acetyl groups. The mass spectral data of phloroglucinol derivatives showed a series of fragment ions apparently produced by gradual losses of hydroxyl, acetyl and methoxy, and five-membered-ring ion resulted from benzene ring cracking.

By comparing with mass spectra of a reference standard, compounds 1, 8 and 10 were identified as 2,4-dihydroxy-6-methoxy-3-methylacetophenone, 2,2,4,4′-tetrahydroxy-6,6′-dimethoxy-3,3′-dimethyl-7,5′-bisacacetophenone and 3,3′-diacetyl-1,4,4′-dimethoxy-2,2′,6,6′-tetrahydroxy-diphenylmethane, respectively (Fig. 3).

3.4. Identification of Jolkinolides type of ent-abietaene lactone (compounds 2, 3, 4, 9, 11 and 12)

A number of Jolkinolides type of ent-abietaene lactones have been reported from Euphorbia species [3]. These lactones are tetracyclic diterpenoids, containing hydroxyl, epoxy groups. By comparison to reference standards, compounds 2, 9 and 11 were identified as Yuexiandujisu D, Jolkinolide B and Jolkinolide A, respectively. A proposed pathway of Jolkinolide B was given in Scheme 1.

Compounds 3 gave [M+H]+ at m/z 351. The MS/MS spectrum showed the same cleavage pattern as Yuexiandujisu D (Fig. 4). Thus, compound 18 was tentatively identified as isomers of Yuexiandujisu D, Jolkinolide B and Jolkinolide A [12]. The MS/MS spectra of compounds 4 ([M+H]+ at m/z 333) and 12 ([M+H]+ at m/z 301), showing neutral losses of H2O and CO, were very similar to those of compounds 2, 9 and 11. Compounds 4 and 12 were tentatively identified as ebracteolatanolide A and Jolkinolide B, respectively.

3.5. Identification of other free diterpenes (compounds 5, 6 and 7)

Compounds 5 and 6 gave [M+H]+ at m/z 305 and m/z 317, respectively. The MS/MS spectra of both compounds showed a similar fragmentation pathway with neutral losses of H2O and CO. Compounds 5 and 6 were tentatively identified as Yuexiandujisu F [11] and Yuexiandujisu C [5], respectively. Compound 7 gave [M+H]+ at m/z 319. The MS/MS spectrum showed losses of H2O, CH3 and COOH. Thus, compound 7 was tentatively identified as Yuexiandujisu A [4].

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