The bioactive, antioxidant, antibacterial, and physicochemical properties of a range of commercially available Australian honeys

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ABSTRACT
Honey is a supersaturated sugar solution produced from plant nectar, with its composition influenced by geographic and floral origins, and with several properties contributing to its health-related abilities. This study aimed to determine the bioactive composition, antioxidant characteristics, antibacterial activity, and physicochemical properties of commercial Australian honeys. In total, 42 commercial Australian honeys were selected, and categorised according to front-label descriptions. Honeys were analysed: quality (Hydroxymethylfurfural); colour (colour intensity, L*,a*,b*); bioactive composition (phenolic, flavonoid, and carotenoid content); antioxidant characteristics (DPPH, CUPRAC, FRAP); antibacterial activity (MIC50); physicochemical properties (pH, TSS, viscosity, αw). Colour intensity correlated with each assessed bioactive compound and antioxidant characteristic (p ≤ 0.001). MIC50 (S. aureus) was associated with FRAP and αw, suggesting mechanisms of action for honey’s antibacterial activity. Manuka-type honeys had higher colour intensity (1440 (98.5) mAU) than other categories (p ≤ 0.05), and consistently higher bioactive and antioxidant properties. This provides the potential to inform antioxidant-related health outcomes.

1. Introduction
Honey is used as a food source and for medicinal purposes across a variety of different cultures (Hunter et al., 2020; Mandal and Mandal, 2011), and is produced as a combination of nectars and bees salivary secretions (Codex Alimentarius, 2001). Honey is a supersaturated sugar solution consisting predominantly of fructose (~38%) and glucose (~31%) (Pascual-Mate et al., 2018), and a variety of over 200 other substances (da Silva et al., 2016). An array of these compounds are considered antioxidants, including phenols, flavonoids, carotenoids, and other enzymatic and non-enzymatic compounds (Bueno-Costa et al., 2016; da Silva et al., 2016). Honey’s composition, including antioxidant content, is dependent on its floral source and geographic origins (Kavanagh et al., 2019; Liu et al., 2013), which also influence the appearance of honey’s physicochemical properties (García et al., 2020).

The in vitro physiological and health-related benefits of some of honey’s compounds are well established (Bogdanov et al., 2008). The effects include honey’s antioxidant activity, which can be attributed to honey’s composition, including its bioactive compounds (Naumovski, 2015). Honey’s antioxidant compounds can reduce the effects of oxidative reactions, such as reducing cellular damage caused by free radical production in vitro (Liu et al., 2013). Additionally, the bioactive compounds, notably phenols, have been demonstrated to contribute to the antibacterial activity of honey (Combarros-Fuertes et al., 2020). The antibacterial properties of honey can also be attributed to its physicochemical properties, such as pH and osmolarity (Combarros-Fuertes et al., 2020), and compounds toxic to bacterial cells in vitro, including hydrogen peroxide, and methylglyoxal in Manuka honey (Bang et al., 2003; Deng et al., 2018). Honey is also associated with decreases in inflammation in vitro through the simultaneous effects of its antioxidant and antibacterial properties (Ruiz-Ruiz et al., 2017).

In Australia, honey is sold as a pure product and must not contain additional substances, including added sugars (Food Standards Australia New Zealand, 2015). Commercial honeys are subjected to heating (45 °C
for 8 h) (Chen et al., 2012) and filtration, which contributes to maintaining product quality and consistency (Kortesniemi et al., 2018). However, commercial treatments have potential to negatively alter honey’s physical properties, in addition to its antibacterial potential (Chen et al., 2012). Further, as some antioxidants degrade during thermal processing (Munialo et al., 2019), potential alterations to honey antioxidant quality may also occur. However contradictory evidence identifies that appropriate processing methods can increase honey’s total antioxidant activity (Escrich et al., 2014).

The bioactive, antioxidant, antibacterial, and physicochemical properties of different types of honey have previously been described in literature, considering their locations and botanic origins (Bodó et al., 2020; Dzugan et al., 2020). However, the analysis of Australian commercial honeys is relatively unexplored (Aazza et al., 2014; Beretta et al., 2005; Saxena et al., 2010). In Australia, commercial honeys represent a considerable market proportion, with 70% of honeys commercially available Australian honeys.

### 2. Materials and methods

#### 2.1. Chemicals and reagents

All chemicals used were of analytical grade. Ethanol, Methanol, (±)-6-hydroxy-2,5,7,8-tetramethylchromane-2-carboxylic acid (Trolox), 2,2-Diphenyl-1-picrylhydrazyl (DPPH), 2,4,6-tris(2-pyridyl)d-s-triazine (TPTZ), Acetone, Aluminium Chloride, Ammonium Acetate, Catechin, Copper (II) Chloride Dehydrate, Ferric Chloride, Folin-Ciocalteu Reagent, Gallic Acid, Glacial Acetic Acid, Hexane, Hydrochloric Acid, Neocuprine, Potassium Hexacyanoferrate (II), Sodium Acetate Trihydrate, Sodium Carbonate, Sodium Hydrogen Sulphate, Sodium Hydroxide, Sodium Nitrate, and Zinc Acetate were purchased from Sigma-Aldrich (Castle Hill, Sydney, Australia). Nutrient Agar, Plates and Nutrient Broth were purchased from Thermo Fisher Scientific (Thebarton, South Australia, Australia). Deionised (DI) water was collected following purification (Millipore Australia, North Ryde, NSW, Thebarton, South Australia, Australia). Deionised (DI) water was collected following purification (Millipore Australia, North Ryde, NSW), with a resistivity higher than 18 MΩcm⁻¹.

#### 2.2. Samples

Honey samples (n = 42) were purchased in 2017 from chain supermarkets (Aldi®, Coles®, Independent Grocers Australia®, Woolworths®) across the Australian Capital Territory and surrounding regions based on availability at time of purchase. All samples were stored in darkness at room temperature (26 ± 3°C) as per Beretta et al. (2005). Honeys were then sorted alphabetically and assigned an identification letter ranging from A to AP for reporting. Limited information regarding the floral and geographical origins of the commercial honeys were provided on the label, except for stating the honeys were of Australian origin. As these honey samples were purchased directly from supermarket retailers, only the information provided on the labels was available for use. However, from the front-label packaging information provided, the samples were categorised as: (1) Manuka (including honeys stated to contain Manuka) honey, (2) Organic honey, (3) Generic Brand honey, (4) honeys with specified Floral origins, (5) honeys from a specific Region of Australia, and (6) Pure (otherwise unspecified) honeys.

#### 2.3. Preparation of samples

Honeys were diluted with DI water to predetermined concentrations, with values converted to 100% concentrations following analysis. Diluted samples were sonicated (FXP4, Ultrasonics Australia, Brookvale, NSW) on the highest sonication power until a clear, sediment-free solution was obtained (Beretta et al., 2005). This allowed for the production of an evenly distributed honey-water solution. Samples did not exceed 50°C to prevent potential temperature-related degradations of bioactive compounds (Beretta et al., 2005).

### 2.4. Honey quality analysis

The Hydroxymethylfurfural (HMF) content of the honeys was determined following the White method (White, 1979). A clarified honey sample was compared to a reference standard in which HMF molecules were destroyed by Sodium Bisulphate (0.2% w/v). Both samples were measured spectrophotometrically (GeneQuant 1300, GE Healthcare, Silverwater, NSW, Australia) at 284 nm and 336 nm against a DI water blank, with HMF content reported in mg/kg of honey, and determined by the following:

$$HMF\ (mg/100g\ of\ honey) = \left(\frac{(Abs_{284} - Abs_{336}) \times Factor}{W}\right)$$

(Note: W = weight of sample (grams); Factor = (126 × 100×1000×100)/ (16830×1000) = 74.87 (126 = honey molecular weight; 16830 = molar absorptivity of HMF at 284 nm) (White, 1979)).

#### 2.5. Bioactive composition

##### 2.5.1. Total phenolic content

Total phenolic content (TPC) was determined using the Folin-Ciocalteur method (Kähkönen et al., 1999) in triplicate, modifying for honey analysis. Absorbance was measured at 765 nm against a DI water blank, using a UV spectrophotometer (Multiskan Go, thermo Scientific, MA, USA). Results were expressed as milligram Gallic Acid equivalents (GAE) per gram of honey (mg GAE/g).

##### 2.5.2. Total flavonoid content

Total flavonoid content (TFC) was determined according to Wu and Ng (2008). The absorbance was measured at 510 nm against a DI water blank (Multiskan Go, Thermoscientific, MA, USA), with results expressed as microgram Catechin equivalents (CE) per gram of honey (µg CE/g).

##### 2.5.3. Carotenoid content (β-carotene and lycopene)

Carotenoid equivalents were determined using a modified version of the method described by Ferreira et al. (2009). Initially, 2 mL of each 50% (w/v) honey sample was combined with 10 mL hexane-acetone mixture (6:4), sonicated immediately for 10 min, and filtered (Whatman No. 1 filter paper). Absorbance was determined spectrophotometrically (Novaspec Plus, Visible Spectrophotometer, Amersham Biosciences, UK) at 453 nm, 505 nm, and 663 nm against a DI water blank. The concentrations of β-carotene (β-CE) and lycopene (LE) equivalents were determined using the following equations, with results expressed as milligram of carotenoid per kilogram of honey (mg β-CE/kg; mg LE/kg).

$$\beta - \text{carotene (mg/100mL)} = 0.216 \times Abs_{505} - 0.304 \times Abs_{565} + 0.452 \times Abs_{483}$$

$$\text{Lycopene (mg/100mL)} = 0.0458 \times Abs_{483} + 0.372 \times Abs_{505} + 0.452 \times Abs_{545}$$

#### 2.6. Antioxidant characteristics

##### 2.6.1. 2,2-Diphenyl-1-picrylhydrazyl (DPPH) assay

The antioxidant radical scavenging capacity was measured using the DPPH assay according to Thaipong et al. (2006), with modifications according to honey analysis. Absorbance was measured in triplicate at 515 nm (Multiskan Go, Thermoscientific, MA, USA). Results were

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**References**: Cited in the text and included at the end of the document. This information is crucial for academic integrity and provides the foundation for the scientific discussion. It ensures that the reader can trace back to the original sources for further research and validation of the presented findings.
expressed as millimoles of Trolox equivalents (TE) per gram of honey (mmol TE/g) using the following (note: M Trolox = 250.29 g/mol) 

\[ \text{Trolox equivalent} = \frac{(\text{Inhibition} \% - (\text{intercept} / \text{slope})/M_{\text{Trolox}})*1}{1} \]

The percentage of inhibition of antioxidant activity was determined as follows:

\[ \text{DPPH Inhibition}\% = (1 - (\text{sample Abs} / \text{reagent blank}))\times100 \]

2.6.2. Cupric reducing antioxidant capacity (CUPRAC) assay

The cupric reducing antioxidant capacity (CUPRAC) was determined according to Apak et al. (2004). The absorbance was measured in triplicate at 450 nm (Multiskan Go, Thermo Scientific, MA, USA) and expressed as millimoles of Trolox equivalents per gram of honey (mmol TE/g).

2.6.3. Ferric reducing antioxidant power (FRAP) assay

The FRAP assay was completed according to Thaipong et al. (2006). The absorbance was measured in triplicate at 593 nm (Multiskan Go, Thermo Scientific, MA, USA), with results expressed as millimoles of Trolox equivalents per gram of honey (mmol TE/g).

2.7. Colour analysis

2.7.1. Colour intensity

The colour intensity was measured using the method of Beretta et al. (2005). The 50% (w/v) honey samples were filtered (Whatman No. 1 filter paper), and absorbance was determined spectrophotometrically (Multiskan Go, Thermo Scientific, MA, USA) at 450 nm and 720 nm. The colour intensity is reported as the difference in spectrophotometric absorbance at the two wavelengths against a DI water blank and expressed as mAU.

2.7.2. \(L^*, a^*, b^+\)

The International Commission on Illumination (CIE) \(L^*, a^*, b^+\) colour measurements of the honey samples were determined by evenly distributing 50 g of each sample around a clear plastic Petri dish 8.5 cm in diameter (Bertoccelj et al., 2007). Measurements were taken in quintuplicate against a white coloured background and reported as per the values provided by the equipment (Colour Reader CR-20, Konica Minolta, Japan).

2.8. Antibacterial activity

The antibacterial activity was measured using the minimal inhibitory concentration (MIC\(_{50}\)) method, as per Patton et al. (2006). Serial dilutions (\(n = 13\)) of unfiltered honey were prepared using nutrient broth. Honey-nutrient broth solutions were inoculated with test culture lutions (\(n = 13\)) of unfiltered honey were prepared using nutrient broth.

Honey-nutrient broth solutions were inoculated with test culture solutions (standardised to McFarland 3.0) of Gram-positive Staphylococcus aureus or Gram-negative Pseudomonas aeruginosa to give 5% (v/v) concentrations, mixed, and aliquoted into wells. Immediately following this, spectrophotometric absorbance was determined in triplicate at 620 nm (\(T_0\)) (Multiskan Go, Thermo Scientific, MA, USA) against a control well containing inoculated broth without honey. Samples were incubated in darkness for 24 h at 37 °C, and absorbance was measured again at T\(_{24}\) at 620 nm. To calculate the MIC\(_{50}\), absorbance at T\(_0\) was subtracted from T\(_{24}\) with the control well assigned a value of 100% growth. The percent inhibition of bacterial growth by the honey samples was determined using the formula:

\[ \text{Percent Inhibition} = 100 - ((\text{sample Abs} T_{24} - T_{0}) / (\text{control Abs} T_{24} - T_{0}))*100 \]

When the difference between T\(_0\) and T\(_{24}\) recorded a negative value, the percent of bacterial growth was calculated:

\[ \text{Percent Growth} = ((\text{sample Abs} T_{24} - T_{0}) / (\text{control Abs} T_{24} - T_{0}))*100 \]

The MIC\(_{50}\) was calculated using GraphPad Prism version 9 (San Diego, California, USA), and is reported as the concentration of honey (w/v) required to inhibit 50% of bacterial growth.

2.9. Physicochemical properties

2.9.1. pH

The pH of the undiluted honey samples was determined in triplicate using a pH meter (Mettler Toledo, Port Melbourne, Australia) at room temperature (22.5 °C ± 0.660) (Ameeruddu et al., 2019).

2.9.2. Total soluble solids

The total soluble solids (TSS) were determined using a handheld digital refractometer (Opti-Brix 54, Bellingham – Stanely, Kent, United Kingdom) (Chan et al., 2017), modified for the use of 50% honey dilutions to allow for equipment requirements. Measurements were completed in triplicate and are expressed as Brix.

2.9.3. Viscosity

The viscosity of the undiluted honey samples (Pa s) was determined using a viscometer (Smart Series, Fungilab, Barcelona, Spain) using an R6 spindle at 5, 10 or 20 rpm, depending on the % torque of the sample (Yücel & Sultanoglu, 2013).

2.9.4. Water activity (a\(_w\))

The a\(_w\) of the undiluted honey samples was determined at room temperature (24.6 °C ± 0.702) using an a\(_w\) measuring instrument (LabSwift-aw, Novasina, Switzerland) according to manufacturer’s instructions.

2.10. Statistical analysis

Statistical analysis was completed using IBM SPSS Statistics version 25 (Armonk, NY: IBM Corp). Variables were examined to determine suitability for parametric or non-parametric methods using histograms, and the Shapiro-Wilk test for normality. Normally distributed variables are presented as mean ± standard deviation, while not normally distributed measures are presented as median (interquartile range). Due to the inclusion of not normally distributed variables, we utilised Kendall’s Tau coefficient of correlation to determine relationships between each of the parameters. To identify the influence of each variable of honey, a partial correlation was also completed. A One-Way ANOVA or Kruskall-Wallis ANOVA with Bonferroni correction as post hoc analysis was completed depending on the distribution of variables, to determine the differences in honey’s properties between all pre-determined categories. Significance was accepted at \(p \leq 0.05\).

3. Results and discussion

3.1. Honey quality analysis

The formation of HMF results from sugar molecule dehydration in honey during the Maillard reaction that occurs during thermal treatments or storage (Pasisas et al., 2017). The presence of HMF in honey is indicative of degradation and overall loss of honey quality (Onür et al., 2017), and has been identified to be potentially carcinogenic in vitro (Capuano and Fogliano, 2011). Therefore, the restriction of this compound is essential and should not exceed 40 mg/kg (Codex Alimentarius, 2001). All honey samples contained a HMF content lower than internationally recommended levels, with HMF levels ranging from 0.150 (0.150) mg/kg (honey S) to 21.0 (1.50) mg/kg (honey Y). While excessive degradation of the honey samples (Codex Alimentarius, 2001) was not observed, storage duration prior to sample purchase is unknown, potentially contributing to HMF production (Onür et al., 2017).
3.2. Bioactive compounds

A summary of the total phenolic content (TPC) of the commercially available Australian honey samples is presented in Table 1 (complete data presented in Supplementary Table S1). For the analysed samples, the highest reported TPC was 0.663 (0.078) mg GAE (honey Y), with the lowest polyphenolic profile being 0.309 (0.125) mg GAE (honey AB). Similar phenol levels were reported in other international honey studies where Liu et al. (2013) reported TPC of honey ranging from 0.307 ± 0.01 to 0.822 ± 0.03 mg GAE/g. In contrast, Aazza et al. (2014) reported a range of 136.82 ± 0.01 to 0.822 ± 0.03 mg GAE/g. For the analysed samples, honeys could be due to the difference in geographical flora utilised in the production of commercial Moroccan honeys. The wider range of phenolic content investigated within this study could be due to a variety of factors, including the honey’s floral source, in addition to potential quality degradation resulting from commercial treatments.

A specific group of polyphenol includes flavonoids, and the TFC values of these honey samples are shown in Table 1 and Table S1. Honey sample Z (96.4 (7.37) μg CE) contained the greatest amount of the compound, and in comparison, sample A (41.6 (1.48) μg CE) had the lowest amount. The results obtained in this study are similar to findings in previously published literature, with total flavonoid content ranging from 27.07 ± 0.35 to 71.78 ± 0.84 mg CE/kg (Khaliﬁ et al., 2012). The variation observed in both the TPC and TFC of different types of honey investigated within this study could be due to a variety of factors, including the honey’s floral source, in addition to potential quality degradation resulting from commercial treatments.

The carotenoid content of the samples was determined using β-carotene and lycopene equivalents (Ferreira et al., 2009) (Table 1; Table S1). Honey sample Z (69.6 (4.95) mg β-CE/kg) contained the highest β-carotene equivalents, with honey A (17.1 (4.58) mg β-CE/kg)

Table 1

Summary of the bioactive composition, antioxidant characteristics, colour analysis, antibacterial activity, and physicochemical properties of a selection of commercially available Australian honeys.

| Highest Ranked Honeys | Lowest Ranked Honeys |
|-----------------------|----------------------|
| **Bioactive Composition** |                       |
| TPC (mg GAE)a | y1 0.663 (0.078)  
                z1 0.635 (0.049)  
                t4 0.604 (0.124)  
                AB8 0.309 (0.125)  
                AO9 0.344 (0.025)  
                A2 0.360 (0.019) |
| TFC (μg CE)a | Z2 96.4 (7.37)  
                y1 92.5 (12.1)  
                s4 84.6 (11.8)  
                A3 41.6 (1.48)  
                B4 44.9 (2.38)  
                G6 47.0 (4.70) |
| β-carotene (mg/kg)b | Z2 69.6 (4.95)  
                      y1 69.5 (17.9)  
                      x1 51.5 (3.68)  
                      A3 17.1 (4.58)  
                      K5 17.4 (2.32)  
                      Q6 19.3 (15.6) |
| Lycopene (mg/kg)c | Z2 24.9 (17.8)  
                   y1 24.8 (10.0)  
                   z2 23.5 (3.76)  
                   A3 7.4 (3.62)  
                   K5 7.74 (2.60)  
                   AP7 9.78 (1.79) |
| **Antioxidant Characteristics** |                       |
| DPPH (mmol TE) | y1 0.713 ± 0.041  
                 AN2 0.600 ± 0.088  
                 Al4 0.596 ± 0.046  
                 AO6 0.207 ± 0.045  
                 J4 0.242 ± 0.020  
                 H4 0.315 ± 0.021 |
| DPPH Inhibition (%) | 69.0 ± 3.97  
                       58.1 ± 8.46  
                       57.8 ± 4.38  
                       20.5 ± 4.32  
                       23.9 ± 1.94  
                       30.8 ± 2.03 |
| CUPRAC (mmol TE)d | y1 21.8 (1.30)  
                    Z2 20.8 (0.910)  
                    O1 19.8 (0.803)  
                    J4 8.28 (1.31)  
                    AO7 8.51 (0.368)  
                    AJ5 10.4 (0.151) |
| FRAP (mmol TE) | y1 3.76 ± 0.158  
                 O1 3.08 ± 0.635  
                 Z2 2.84 ± 0.266  
                 0.670 ± 0.390  
                 0.841 ± 0.297  
                 1.05 ± 0.168 |
| **Colour Analysis** |                       |
| Colour Intensity (mAU)e | Z2 1440 (98.5)  
                          y1 1196 (42.9)  
                          x1 905 (77.5)  
                          G6 241 (5.30)  
                          k5 261 (4.00)  
                          q6 303 (6.70) |
| L*a | O1 33.6 (0.800)  
      A2 25.1 (1.20)  
      B8 23.8 (1.90)  
      y1 11.7 (0.800)  
      z1 12.0 (0.800)  
      AP3 12.6 (0.850) |
| a* | O1 9.70 ± 0.212  
    AA2 7.50 ± 0.436  
    An2 7.36 ± 1.78  
    AO6 0.980 ± 0.327  
    AO5 1.98 ± 0.455  
    AO3 2.39 ± 0.596 |
| b* | O1 24.7 (0.400)  
    A2 18.2 (2.05)  
    P4 16.7 (3.10)  
    y1 5.30 (0.800)  
    6.00 (1.05)  
    6.20 (0.900) |
| **Antibacterial Activity** |                       |
| S. aureus MIC50 (%) dilution; w/v | L6 5.82  
                                   P6 4.76  
                                   AK2 4.75  
                                   Q6 0.078b  
                                   R6 0.122  
                                   AM6 0.124a |
| **Physicochemical Properties** |                       |
| pHf | A1 4.64 (0.160)  
      Alp 4.60 (0.200)  
      Plp 4.43 (0.170)  
      AP2 3.74 (0.020)  
      AP3 3.81 (0.050)  
      AP4 3.82 (0.160) |
| TSS (°Brix) | H2 60.2 ± 0.001  
              B2 59.3 ± 0.115  
              E2 59.3 ± 0.115  
              AC6 55.3 ± 0.115  
              AB6 55.8 ± 0.001  
              AN6 56.2 ± 0.001 |
| Viscosity (Pa s)g | O2 158.0 (1.32)  
                   H2 76.4 (2.14)  
                   A2 65.0 (26.1)  
                   P2 11.0 (0.275)11  
                   C2 12.8 (0.362)11  
                   AK2 16.8 (0.776)11 |
| a*0.5 | A2 0.583  
       B2 0.572  
       C2 0.568  
       AO2 0.454  
       AN2 0.459  
       AM2 0.461 |

Note: The three highest and lowest ranked honeys are presented. MIC50 was calculated from triplicate data; L*, a*, b* measurements were taken in quintuplicates, a*0.5 was analysed once for each sample, and all remaining variables were completed in triplicate. Normally distributed variables are presented as mean ± standard deviation, and not normally distributed variables are identified by superscripts next to the honey identification letter are related to front-label category descriptions of the honey: 1 = Manuka Honey; 2 = Organic Honey; 3 = Generic Brand Honey; 4 = Australian Floral Honey; 5 = Regional Honey; 6 = Pure Honey.
reporting the lowest. Additionally, the lycopene equivalents range from honey R \((24.9\ (17.8)\ \text{mg\ LE/kg})\) reporting the highest, to honey A \((7.43\ (3.62)\ \text{mg\ LE/kg})\). The composition of carotenoids in honey is dependent on a variety of factors, including the floral and geographic origins (Bueno-Costa et al., 2016), which could be the cause of large ranges observed in the β-carotene and lycopene values.

### 3.3. Antioxidant characteristics

The DPPH assay was utilised to determine the antiradical scavenging capacity of the samples, and both the Trolox equivalent and percentage of inhibition were determined (Table 1; Table S1). It was revealed that honey sample possessing the greatest radical scavenging capacity was Y \((0.713\ ±\ 0.041\ \text{mmol\ TE})\; (69.0\%\ ±\ 3.97)\). Alternatively, honey AE \((0.207\ ±\ 0.045\ \text{mmol\ TE})\; (20.5\%\ ±\ 4.32)\) presented the lowest scavenging ability. Values derived in this study are comparable to other commercial samples, where inhibition was reported between \(1.95\%\ ±\ 0.87–19.12\%\ ±\ 1.34\) (Devarajan and Venugopal, 2012), with Australian samples presenting greater antioxidant potential.

The specific determination of the antioxidant reducing ability of the samples was completed through the CUPRAC assay (Table 1; Table S1), which assesses the ability of antioxidant compounds to reduce the Copper (II) Cupric ion (Apak et al., 2004). Honey sample Y \((21.8\ (1.36)\ \text{mmol\ TE})\) showed the greatest cupric ion reducing ability, while sample J \((8.28\ (1.31)\ \text{mmol\ TE})\) demonstrated the weakest ability. Previous findings demonstrate lower honey CUPRAC values \((124.8\ ±\ 63–532\ ±\ 19\ \mu\text{mol\ TE/g})\) (Ulusoy et al., 2010), which may indicate a higher mineral content within honey analysed in this research, contributing to a higher antioxidant reducing ability of the Australian commercial samples.

The FRAP methodology determines the reducing power of honey by determining the ability of antioxidants present in samples to reduce the molecules \(\text{Fe}^{3+}/\text{Fe}^{2+}\) (Petretto et al., 2015). The greatest ferric reducing ability (Table 1; Table S1) was identified in sample Y \((3.76\ ±\ 0.158\ \text{mmol\ TE})\), with sample J \((0.670\ ±\ 0.390\ \text{mmol\ TE})\) reporting the weakest power. Similar values were recorded for unifloral honeys, with a mean range of 1033.33–2779.17 mmol TE/kg (Sowa et al., 2017).

### 3.4. Relationships between bioactive compounds and antioxidant characteristics

The completion of a Kendall’s Tau correlation identified relationships between each analysed variable (Table 2). The TPC and TFC values showed a significant correlation \((\tau\ =\ 0.566,\ p\ ≤\ 0.001)\), and have also reported positive correlations with each antioxidant characteristic assay \((p\ ≤\ 0.001)\). However, it was observed that when TPC was controlled for, TFC and each antioxidant characteristic were no longer associated \((p\ ≥\ 0.05)\). In contrast, when TFC values were controlled, the antioxidant characteristics remain associated with TFC \((p\ ≤\ 0.05)\), suggesting honey’s TFC content does not significantly contribute to honey’s phenolic content, or the antioxidant potential of phenolic content. This is supported by the TFC for these samples being between 10.1% and 17.9% of the total polyphenolic content, with flavonoid content reported to be ~12.85% of the total polyphenols of honey (Blasa et al., 2006).

The β-carotene and lycopene carotenoid equivalents were positively correlated \((\tau\ =\ 0.580,\ p\ ≤\ 0.001)\), and were associated with both TPC and TFC (β-carotene: \(p\ ≤\ 0.001;\) lycopene: \(p\ ≤\ 0.05)\). While strong associations were observed for β-carotene and the antioxidant characteristics \((p\ ≤\ 0.001)\), lycopene did not correlate with DPPH \((p\ =\ 0.070)\), and weakly correlated with CUPRAC and FRAP \((p\ ≤\ 0.05)\). When controlling for both TPC and TFC, neither β-carotene or lycopene correlated with any of the antioxidant characteristics \((p\ ≥\ 0.05)\), suggesting these carotenoid equivalents do not contribute to honey’s antioxidant potential in isolation.

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**Table 2**

| Kendall’s Tau correlations between the bioactive content, antioxidant characteristics, colour, antibacterial activity, and physicochemical properties of a selection of commercially available Australian honeys. |
|---|---|---|---|---|---|---|---|
| TPC | TFC | β-Carotene | Lycopene | DPPH | CUPRAC | FRAP | Colour Intensity |
|---|---|---|---|---|---|---|---|
| TPC | 1 | 0.566* | 0.700* | 0.670* | 0.575* | 0.515* | 0.596* |
| TFC | 0.566* | 1 | 0.508* | 0.501* | 0.508* | 0.489* | 0.566* | 0.547* |
| β-Carotene | 0.494* | 0.501* | 1 | 0.494* | 0.494* | 0.494* | 0.600* | 0.538* |
| Lycopene | 0.501* | 0.508* | 0.494* | 1 | 0.494* | 0.494* | 0.494* | 0.494* |
| DPPH | 0.700* | 0.508* | 0.494* | 0.494* | 1 | 0.417* | 0.417* | 0.417* |
| CUPRAC | 0.670* | 0.508* | 0.494* | 0.494* | 0.417* | 1 | 0.417* | 0.417* |
| FRAP | 0.575* | 0.508* | 0.494* | 0.494* | 0.489* | 0.489* | 1 | 0.489* |
| Colour Intensity | 0.596* | 0.566* | 0.575* | 0.575* | 0.566* | 0.566* | 0.566* | 1 |

Note: * Correlation is significant at 0.01; ** Correlation is significant at 0.05.
3.5. Colour analysis

Honey’s colour is predominantly related to its pigmented contents, such as flavonoids and carotenoids, which are determined by honeys’ floral origins (Bertonečelj et al., 2007). Honey samples with more pigmented compounds appear darker, in comparison to honeys with lower levels of these compounds (Bueno-Costa et al., 2016). Of the included samples (Table 1; complete data presented in Supplementary Table S2), honey Z (1440 (98.5) mAU) possessed the highest colour intensity, with honey G (241 (5.30) mAU) reporting the lowest. These findings are contradicted by Roshan et al. (2017) who determined a lower range of colour intensity of 41–471 mAU for their selection of unifloral Australian honeys. Observed colour variations of the commercial samples may be due to either floral variation of Australian vegetation, or commercial treatments (Segato et al., 2019).

The lightest honey, as determined by \( L^* \) values, was O (33.6 (0.800)), with the darkest honey being Y (11.7 (0.800)) (Table 1; Table S2). Considering that an \( L^* \) value of 100 represents white, the honeys in this study are relatively dark, with similar studies reporting \( L^* \) values of between 42.12 ± 2.27 to 64.40 ± 0.63 (Bertonečelj et al., 2007). Additionally, \( a^* \) values range between 7.90 ± 0.212 (honey O) and 0.980 ± 0.327 (honey G), with \( b^* \) values ranging from 24.7 (0.400) (honey O) to 5.30 (0.800) (honey Y). Positive \( a^* \) and \( b^* \) values reported for all honeys indicate that the samples present as red and yellow in colour (Szabó et al., 2016), while some honeys present a green profile (Kuš et al., 2014).

3.6. Relationships between colour, bioactive compounds, and antioxidant characteristics

An inverse relationship was observed between colour intensity and \( L^* \), confirming the samples’ lightness (Fig. 1h; \( \tau = -0.538, p \leq 0.001 \)), as supported by previous research (Bertonečelj et al., 2007). Similarly, an inverse relationship was observed between colour intensity and \( b^* \) (Fig. 1j; \( \tau = -0.356, p \leq 0.001 \)), with a strong correlation between \( L^* \) and \( b^* \) (\( \tau = 0.728, p \leq 0.001 \)). As positive \( b^* \) values represent honey’s yellowness, these correlations highlight that yellow is characteristic of light-coloured honeys. In contrast, the association between colour intensity and \( a^* \) (Fig. 1i; \( \tau = 0.594, p \leq 0.001 \)) indicates that darker honeys are characterised by red.

Colour intensity positively correlated (\( p \leq 0.01 \)) with each bioactive and antioxidant variable (Table 2, Fig. 1). Carotenoid content of honey reportedly contributes to its colour through providing pigmentation (Bueno-Costa et al., 2016), with Fig. 1c depicting strong correlations between colour intensity and \( \beta \)-carotene content (\( \tau = 0.600, p \leq 0.001 \)). However, weak correlations between colour and lycopene were observed (Fig. 1d; \( \tau = 0.287, p \leq 0.01 \)). The significant correlation between colour intensity and TPC, TFC, DPHH, CUPRAC, and FRAP (Fig. 1; all \( p \leq 0.001 \)) indicates increases in honey colour relates to increases in antioxidant scavenging capacity, which is supported by the inverse correlation between \( L^* \) values and all bioactive and antioxidant characteristics (\( p \leq 0.01 \)). Previously, pale-coloured honeys were reported to have a lower antioxidant activity than darker variations (Ciompiou et al., 2013; Saxena et al., 2010).

The relationship between colour intensity, bioactive content, and antioxidant characteristics suggests that the darker the honey, the greater its antioxidant potential. Consumers consider the health benefits and nutritive value of honey when selecting honey products (Arvanitoyannis and Krystallis, 2006), including Australian consumers (Batt and Liu, 2012). Therefore, based on our findings and consumer values, there is potential for the utilisation of the relationship between colour intensity and the antioxidant characteristics to increase marketability of honey products. However, in this case, it should be considered that the completion of commercial treatments may also result in the darkening of honeys (Segato et al., 2019).

3.7. Antibacterial activity

The antibacterial activity of the samples was determined through the MIC\(_{50}\) method, where the concentration of honey (% dilution; w/v) required to inhibit 50% of bacterial growth was determined for \( S. aureus \) and \( Ps. aeruginosa \). All samples were effective as an inhibitory agent against \( S. aureus \) (Table 1; Table S2; Fig. 2; complete figure in Supplementary Fig. S1). The most effective MIC\(_{50}\) concentration was 0.078% (Honey Q; Fig. 2a), with the weakest for this species being 5.82% (Honey L; Fig. 2b). Inhibitory activity of honey against this bacterial species is consistently supported by similar research (Mahmoodi-Khaledi et al., 2017; Patton et al., 2006). Bucekova et al. (2020) demonstrated the MIC of commercially available honey samples purchased in Slovakia against \( S. aureus \). While their data cannot be compared to the MIC\(_{50}\) values of the Australian samples, they concluded that the antibacterial activity of commercial honey is not uniform (Bucekova et al., 2020), which is reflected by the MIC\(_{50}\) values presented in this study.

All honeys effectively inhibited \( Ps. aeruginosa \) growth at 50% (w/v) concentrations. However, only 9 honeys (21%) inhibited any growth at 25% (w/v) concentrations (hones identified in Table 1; Table S2), with no further dilutions observed to inhibit growth. Interestingly, these were not the same honeys to demonstrate the lowest MIC\(_{50}\) values for the inhibition of \( S. aureus \), indicating differences in honey effectiveness based on bacterial species. Therefore, MIC\(_{50}\) values could not be determined, with the analysed commercial samples not being an effective antibacterial agent against \( Ps. aeruginosa \). It was previously determined that \( Ps. aeruginosa \) demonstrates antibiotic resistance to honey at weaker concentrations (Mahmoodi-Khaledi et al., 2017), which could be attributed to its gram-negative structure (Breijyeh et al., 2020).

3.8. Mechanisms for honey’s antibacterial activity

3.8.1. Physicochemical properties

The pH of honey is typically accepted as between 3.2 and 4.5 (da Silva et al., 2016), contributing to antimicrobial activity by creating an undesirable pathogenic environment (Lund et al., 2020). Results indicated (Table 1; Table S2) the lowest pH from these samples was 3.74 (0.020) (honey AF), with the highest being 4.64 (0.160) (honey A). The pH was found to be positively associated with TPC and CUPRAC (\( p \leq 0.05 \)), suggesting that honey’s bioactive composition is responsible for its acidity levels, and therefore the pH contribution to antibacterial activity.

The TSS of honey often indicates the sugar compounds present, which contributes to low moisture content and osmotic stress for select microorganisms (Combarros-Fuertes et al., 2020). Of the assessed honeys (Table 1; Table S2), honey H (60.2 ± 0.001 °Brix) reported the greatest TSS content, while honey AC (55.3 ± 0.115 °Brix) reported the least. Our values are lower than previous findings, where soluble solid levels ranged from 76.3 to 85.3 °Brix (Oroian and Ropciuc, 2017). Differences in soluble solid content could be ascribed to different analysis techniques, or as a consequence of commercial treatments applied to our samples (Braghini et al., 2020).

There is a large variation observed for the honey’s viscosity (Table 1; Table S2), with a range of 11.0 (0.275) Pa s (honey P) to 158 (1.32) Pa s (honey O), in addition to a range of rotation speeds (5 rpm, 10 rpm and 20 rpm) required due to the varied consistency. It is established that factors influencing honey’s viscosity are its high sugar concentration and low moisture content (Jiang et al., 2020; Václav & Sultanoglu, 2013). Interestingly, no associations (\( p \geq 0.05 \)) were observed between sample viscosity and TSS, representative of the sugar content, or \( a_n \), which indicates honey’s moisture.

The \( a_n \) of honey is affected by factors such as sugar composition, degree of crystallisation, and water-sugar interactions, and typically measures below 0.60, preventing microorganism growth (Subbiah et al., 2020). The analysed honeys reported a lower \( a_n \) ranging from 0.454 (honey AO) to 0.583 (honey A) (Table 1; Table S2). The sample \( a_n \) was
Fig. 1. Correlation between Colour Intensity (mAU) and (a) Total Phenolic Content (mg GAE), (b) Total Flavonoid Content (μg CE), (c) β-carotene equivalents (mg/100 mL), (d) Lycopene equivalents (mg/100 mL), (e) DPPH (mmol TE), (f) CUPRAC (mmol TE), (g) FRAP (mmol TE), (h) L*, (i) a*, (j) b*, of a selection of commercially available Australian honey samples. Statistical significance is p ≤ 0.01.
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3.8.2. Relationships between antibacterial activity and antioxidant characteristics

The MIC\textsubscript{50} values of the assessed honeys against \textit{S. aureus} also correlated with FRAP (τ = 0.220; p ≤ 0.05) (Table 2), indicating the contribution of honey’s antioxidant compounds to its antibacterial activity. This is further supported by Leyva-Jimenez et al. (2019), where the extracted phenolic content of honey had similar antibacterial potency as whole honey samples for a range of bacterial species (Leyva-Jimenez et al., 2019). Despite no correlation being found between MIC\textsubscript{50} associated with the TSS (τ = 0.237; p ≤ 0.05) (Table 2), however this was not anticipated due to the influence of sugar content on the water activity (Oroian and Ropciuc, 2017). The \( a_w \) was also associated with the MIC\textsubscript{50} for \textit{S. aureus} (τ = 0.222; p ≤ 0.05) (Table 2). Interestingly, other assessed physicochemical properties did not correlate with MIC\textsubscript{50} (p ≥ 0.05) despite reports of these properties contributing to honey’s antibacterial activity (Combarros-Fuertes et al., 2020; Mandal and Mandal, 2011).

Fig. 2. The MIC\textsubscript{50} (% w/v) of a selection of commercially available Australian honeys against \textit{S. aureus}; (a) represents the four lowest honey concentrations recorded of the studied samples, (b) represents the four highest honey concentrations recorded.
and TPC, when the phenolic content is controlled for, the MIC\textsubscript{50} of the samples is no longer associated with FRAP (p = 0.198). This change in significance highlights the contribution of the phenolic content to honey’s antibacterial activity.

### 3.9. Characteristics of honey categories

Manuka samples (Category 1) reported among the largest means for each antioxidant assay, indicating samples in this category had the highest antioxidant capacity. Specifically, honeys Y and Z frequently reported among the highest for each bioactive (TFC, TPC, β-carotene, lycopene) and antioxidant (DPPH, CUPRAC, FRAP) (Table 1) method. In contrast, Generic Brand (Category 3) and Australian Floral honeys (Category 4) reported the lowest values for nearly all parameters. It was anticipated that Manuka honeys would possess the greatest antibacterial activity, based on previous effectiveness compared with other honey types, particularly for S. aureus (Gosliński et al., 2020). However, in our study, a Pure (Category 6) and Regional (Category 5) honey (Table 1) reported the highest antibacterial ability. For the physicochemical properties, no categories consistently reported the highest or lowest values, indicating that honey type does not influence these factors.

The difference between each category was determined for each of the assays using One-Way ANOVA or Kruskal-Wallis, depending on the normality, with differences observed only between Manuka honeys and some other categories. These were observed for FRAP between Manuka and Generic Branded (p = 0.001), Australian Floral (p = 0.001), Regional (p = 0.009) and Pure (p = 0.02) honeys. Similarly, TFC was significantly higher in Manuka honey than Generic Branded (p = 0.001), Australian Floral (p = 0.001), and Pure (p = 0.002) honeys, and Manuka honey had a greater colour intensity than Australian Floral honeys (p = 0.003).

Many studies that determine the antioxidant and antibacterial properties of honey report distinct floral and geographic origins of their samples. These studies have reported the antioxidant characteristics, specifically the phenolic composition, is due to these factors (Escuredo et al., 2013; M. I. Khalil et al., 2011). However, this information is not available for commercial samples unless provided by the manufacturer on the packaging. Despite this, the honeys investigated in this study are representative of what is available to consumers on the Australian commercial market.

### 3.10. Future directions and limitations

To our knowledge, this is the first study examining the bioactive, antioxidant, antibacterial, and physicochemical properties of commercially available Australian honeys. Although consumer behaviour towards Australian honeys, including commercial honeys, has been identified (Batt and Liu, 2012), there has been no analysis determining which sensory values drive likeability and purchasing behaviour. Based on the observed antioxidant characteristics of the honeys, there is potential for further investigation into antioxidant-related health benefits of commercial Australian honey, including determining if these samples are efficient at reducing plasma antioxidant content in vivo. Additionally, given the observed association between the antioxidant characteristic FRAP and antibacterial activity, the synergistic anti-inflammatory potential should be investigated (Hunter et al., 2020).

The findings of this investigation are limited by the techniques used, with spectrophotometric analysis used for the quantification of bioactive and antioxidant compounds. The technique measures the refractive light absorption of solid compounds, and often detects different compounds at the same wavelength, resulting in an overestimation of content (Biehler et al., 2010). Further, an analysis of the polyphenolic profile of the samples was not completed, which could improve this research by allowing for the identification of compounds that most contribute to the sample’s antioxidant characteristics. Finally, it has been reported that a considerable contributing variable to honey’s antibacterial ability is the presence of hydrogen peroxide (Bang et al., 2003), however, this was not assessed, with its contribution to the MIC\textsubscript{50} of the samples unknown.

### 4. Conclusions

In conclusion, the colour intensity of the honeys is related to a variety of bioactive properties and antioxidant characteristics. Additionally, the antibacterial activity was associated with the honey’s water activity, and the antioxidant characteristic FRAP. The phenolic content of the samples contributed to both antioxidant and antibacterial properties of the honeys, as demonstrated by changes in significance when controlling for this variable. From all honeys tested in this study, Manuka honeys consistently reported higher results for some of the included assays, with a variety of generic branded and specific floral honeys reporting the lowest values. These findings can potentially inform honey producers and consumers on future directions of honey marketability through highlighting the potential antioxidant and antibacterial-related health benefits.

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### CRediT authorship contribution statement

Maddison Hunter: Conceptualization, Methodology, Validation, Investigation, Formal analysis, Visualization, Writing – original draft, Visualization, Writing – review & editing. Reena Ghildiyal: Methodology, Validation, Formal analysis, Writing – review & editing. Nathan M. D’Ginha: Validation, Writing – review & editing. Caroline Gouw: Validation, Writing – review & editing. Ekavi N. Georgouposoulosou: Formal analysis, Writing – review & editing. Nenad Naumovski: Conceptualization, Methodology, Visualization, Writing – review & editing, Supervision.

### Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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### Appendix A. Supplementary data

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