The Microstructure and Mechanical Properties of Cu-20Ni-20Mn Alloy Fabricated by a Compact Preparation Process

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Abstract: A novel compact preparation process has been developed to produce a Cu-20Ni-20Mn alloy. This process involves heating-cooling combined mould (HCCM) continuous casting, a solution treatment at 800 °C, rolling at room temperature and a final ageing step at 450 °C. This process eliminates two hot deformation processes, namely, hot forging and hot rolling, greatly improving production efficiency and reducing production costs compared with the traditional preparation process. The alloy fabricated by this process was found to have excellent mechanical properties. Additionally, the formation of precipitated phases during the ageing step is accelerated by this new process. The hardness of the samples reaches 476 HV after ageing for 10 h. Stable, ordered precipitates of Ni3Mn(L12 phase) are observed in the rolled specimen, and the orientation relationship between the copper matrix and ordered Ni3Mn phase (L12 phase) is [200]Cu//[010]Ni3Mn and [011]Cu//[011]Ni3Mn. Precipitation strengthening is the main reason for the increase of strength of the sample during the ageing process. The mechanical properties and ageing precipitation process of the alloy are affected by the rolling process. The precipitation rate of the rolled sample is greatly increased during the ageing process, leading to a larger amount of precipitates.

Keywords: Cu-20Ni-20Mn alloy; compact preparation process; mechanical properties; microstructure evolution; combined ageing process

1. Introduction

High-strength elastic copper alloys have been widely used in many fields and applications, such as the aerospace field and in petrochemical electronic equipment, due to their excellent properties, namely, their high strength, elasticity, good thermal conductivity, stress relaxation resistance, excellent corrosion resistance and wear resistance [1–4]. The Cu-Be alloy is considered as an ideal elastic material for applications because it exhibits comprehensive properties that are better than most other copper alloys [5]. However, the production and application of Cu-Be alloys are severely restricted [6,7] due to their need for harsh processing conditions and high production cost. In addition, beryllium oxide and compounds produced in the melting and processing processes are extremely harmful to the human body and environment.

Cu-20Ni-20Mn alloy can be used to replace beryllium copper in some areas. Cu-20Ni-20Mn alloy have a higher strength and elastic modulus than Cu-Be alloys, along with being very stable at
This paper proposed a compact process consisting of HCCM continuous casting [13], a solution process is observed by XRD, SEM and TEM techniques. The influence of the rolling deformation on the structure and properties of Cu-20Ni-20Mn alloy was carried out by conducting a series of ageing precipitation tests of specimens rolled at room temperature. The phase transformation sequence during the ageing process is observed by XRD, SEM and TEM techniques. The influence of the rolling deformation on the structure and properties of Cu-20Ni-20Mn alloy is analysed. Finally, the strengthening mechanisms that correspond to precipitation behaviours are discussed to provide a theoretical direction for the optimization of this alloy.

2. Experimental

2.1. Material Preparation

The Cu-20Ni-20Mn (mass fraction, %) alloy was prepared using pure Cu, CuNi20 master alloy and CuMn22 master alloy by HCCM continuous casting technology. Figure 1 shows a schematic of the principles of HCCM continuous casting technology.

![Figure 1](image.png)

**Figure 1.** Principle schematic of the HCCM continuous casting technology: 1—graphite stopper, 2, 7—thermocouples, 3—graphite crucible, 4—temperature measuring device, 5—alloy solution, 6—graphite thermal type, 8—graphite mold, 9—dummy bar head, 10—crystallizer, 11—tow bar.

The main feature of the HCCM continuous casting is that the mold is composed of a heating section and a cooling section. The mold heating section is heated by induction coils and the heating temperature is higher than the melting point of the alloy. A forced cooling is performed at the cooling section with a water-cooling copper sleeve. The processing parameters used in the present work were listed as follows: melting and holding temperature 1250 °C, cooling water flow rate 600 L/h, drawing speed 60 mm/min. The shape of the casting depends on the shape of the mold and the mold can be
designated according to the needs to obtain samples of different shapes. The alloy prepared by HCCM continuous casting technology could obtain excellent performance after combined ageing treatment. The combined ageing processing parameters used in the present work were listed as follows: solution treatment at 800 °C for 1 h, rolling at room temperature with 75%, 85% or 95% reduction, and final ageing at 450 °C for different time.

2.2. Mechanical Properties

Hardness tests were conducted on a HXD-1000TM/LCD Vicker’s hardness tester (Shanghai optical instrument factory, Shanghai, China), with a holding time of 15 s and a load of 3 kg. The tests corresponding to each condition were repeated at least five times to minimize experimental error.

2.3. Microstructure Observations and Analysis

The samples for SEM were first ground, polished and etched in a mixed solution containing 10 g FeCl$_3$ + 30 mL HCl + 120 mL C$_2$H$_5$OH. The specimens for the TEM observations were prepared by a focused ion beam (FIB) with a size of 30–60 nm, which was beneficial for observing by TEM. The microstructures of the samples were observed by scanning electron microscopy (SEM, JEOL-7900 field emission scanning electron microscope, JEOL, Tokyo, Japan) and transmission electron microscopy (TEM, JEM2100 transmission electron microscope, JEOL, Tokyo, Japan).

The specimens for X-ray diffraction, with dimensions of 15 mm × 25 mm × 0.3 mm, were ground with 1500 grade water-proof abrasive paper. The experiments were carried out using X-ray diffraction (XRD, D8 Advance X-ray diffractometer Bruker, Billerica, MA, USA) with a CuKα radiation source at an accelerating voltage of 40 kV, a current of 40 mA and a continuous scanning rate of 4°/min.

3. Results

3.1. Variation of the Hardness of the Samples Treated with Different Process

The variations in the hardness of the solid solution specimens treated with different processes are shown in Figure 2. The hardness of the solid solution specimen is 125 HV. There is no significant change in the hardness of the solid solution specimen at the early stage of ageing. The hardness of the solid solution specimen increases rapidly after ageing for 2 h, with its value reaching up to 400 HV after ageing for 24 h. The hardness of the solid solution specimens after ageing for 24 h is approximately the same.

![Figure 2. Variation of hardness of the specimens during aging process, (a) the variation of hardness of solid solution specimen, (b) the variation of hardness of specimen treated with different reduction.](image-url)
respectively. The hardness evolutions of the specimens rolled with different reductions during the ageing process are approximately the same. The hardness values of the rolled specimens increase rapidly at the early stage of ageing. The hardness values of the specimens after ageing for 10 h are approximately the same. Notably, the hardness of the specimen rolled with a 95% reduction reaches a hardness of 476 HV after ageing for 10 h. The hardness of the specimen fabricated by traditional preparation process is 487 HV in previous study [14]. The hardness of the specimen fabricated by novel compact process is approximately the same as the hardness of the specimen fabricated by traditional preparation process.

These results indicate that the rolling process is beneficial for increasing the hardness of a specimen. Furthermore, the hardness of a sample is affected by the reduction during rolling. The hardness of a sample increases with an increase in the reduction value. The evolution of hardness during the ageing process is also influenced by the rolling process. The precipitation rate during the ageing process of the sample becomes faster through the effect of cold deformation. Therefore, the time required to reach the peak value of strength during the ageing process of the rolled samples is greatly reduced compared with that of the solid solution sample because there is an incubation period during the ageing of solid solution samples. However, the incubation period is not found during the ageing process of the rolled samples. In addition, the hardness of the rolled samples at their peak ageing states are higher than that of the solid solution state. These changes in hardness are caused by the evolution of the microstructure, which will be analysed in the following discussion section.

3.2. Microstructure of the Sample Treated with Different Rolling Process

The microstructures of specimens treated with different rolling processes are demonstrated in the SEM micrographs shown in Figure 3. A large number of columnar grains are found in specimens prepared by HCCM continuous casting technology, as shown in Figure 3a. Moreover, these columnar crystals are not eliminated by the solution treatment. Therefore, a large amount of columnar crystal structures are still observed in the solution-treated sample, as shown in Figure 3b. The microstructure of the sample changes through cold deformation. The microstructure of the rolled specimen is found to transform from columnar grains to slender fibrous structures, as seen in Figure 3c–e.

![Figure 3. SEM micrographs of specimens treated with different rolling processes: (a) as-cast specimen, (b) solid solution specimen, (c) rolled specimen with a 75% reduction, (d) rolled specimen with an 85% reduction, and (e) rolled specimen with a 95% reduction.](image-url)

The grain size will be affected by the reduction during rolling. The grain size becomes very fine after cold deformation with a 95% reduction, as shown in Figure 3e. Thus, the mechanical properties
and ageing precipitation process of the alloy will be affected by cold deformation, which will be analysed in the following discussion section.

3.3. Microstructure Evolution of the Rolled Sample during Aging Process

The TEM micrographs and corresponding SAED patterns of the rolled specimens aged at 450 °C for 2, 10 or 48 h are shown in Figure 4. Two kinds of precipitates can be found in the sample, as seen in Figure 4a. One kind is a very small precipitate of approximately 5–10 nm, and the other is approximately 15–20 nm. The diffraction spots of the Ni₃Mn phase (L₁₂ phase) are found in Figure 4b, which have previously not been found in a Cu-20Ni-20Mn alloy. The orientation relationship between the copper matrix and ordered Ni₃Mn phase (L₁₂ phase) is (200)Cu//(010)Ni₃Mn and [011]Cu//[011]Ni₃Mn. A large amount of precipitates can also be found in the rolled specimens aged for 10 h, as seen in Figure 4c. The orientation relationship between the copper matrix and ordered Ni₃Mn phase (L₁₂ phase) is (020)Cu//(010)Ni₃Mn and [001]Cu//[001]Ni₃Mn. Furthermore, the number of precipitates observed in the sample after 48 h of ageing increase, and the sizes of the precipitates observed in the 48 h-aged sample are approximately the same. In addition, many interference fringes are observed in the rolled specimen aged for 48 h, as shown in Figure 4e. The reason for these interference fringes is due to the abundance of small, coherent precipitates that are produced from the matrix.

**Figure 4.** TEM micrographs and SAED patterns of the rolled specimens aged at 450 °C for 2, 10 or 48 h: (a) TEM micrograph and (b) [011]Cu SAED pattern of the rolled specimen aged for 2 h, (c) TEM micrograph and (d) [001]Cu SAED pattern of the rolled specimen aged for 10 h, and (e) TEM micrograph and (f) [001]Cu SAED pattern of the rolled specimen aged for 48 h.
The Ni$_3$Mn phases are still found in the 48 h-aged sample, which is confirmed by the diffraction spots of the ordered Ni$_3$Mn phase (L1$_2$ phase) found in the SAED pattern (Figure 4f). It is difficult to determine whether the NiMn phase is also present in the specimen by the SAED pattern because the position of the partial diffraction spots of the Ni$_3$Mn phase coincides with the diffraction spots of the NiMn phase. The existence of NiMn is discussed later in the XRD analysis. This is the first study to show the Ni$_3$Mn phase in an aged Cu-20Ni-20Mn alloy. The generation of these Ni$_3$Mn phases and their impact on the alloy properties are discussed in the following discussion section.

Figure 5 shows the XRD patterns of the rolled specimens aged at 450 °C for different times in two different 2θ ranges. The (220) peaks of the rolled specimen gradually widen as the ageing time increases, and the peaks of the ordered NiMn phase appear in the vicinity of the matrix peak in the XRD patterns of the rolled specimens aged for a long period of time, as shown in Figure 5. The XRD patterns confirm the existence of the NiMn phase during the ageing process in the rolled samples, indicating that there are two types of precipitated phases (NiMn and Ni$_3$Mn) in the specimens after ageing for 2 h. These results are in good agreement with the TEM results.

![Figure 5. The XRD patterns of the rolled specimens aged at 450 °C for different times in different 2θ ranges, (a) 30–90°, (b) 70–80°.](image)

4. Discussion

4.1. Strengthening Mechanism of the Alloy during the Ageing Process

It can be seen from the previous results that the hardness of the alloy increases greatly through the ageing process. The hardness of the rolled specimen reached 476 HV after ageing for 24 h, while its hardness before the ageing process was only 297 HV. The evolution of the hardness is caused by the evolution of the microstructure. According to the results above, the microstructural evolution of the rolled specimens during the ageing process can be identified. The precipitates consist of two phases (NiMn and Ni$_3$Mn) that rapidly precipitate at the early stage of ageing. The NiMn phase is considered the main precipitation phase during the ageing process [8,15], with L1$_0$ structure. The structure of the Ni$_3$Mn phase with face-centred cubic (L1$_2$ structure) structure is found in the aged Cu-20Ni-20Mn alloy, which is firstly reported in this alloy. The Ni$_3$Mn phase is also an ordered precipitated phase, with the size of 15–20 nm, as seen in Figure 6, which is larger than that of the NiMn phase. The Ni$_3$Mn phase is considered to be transformed from the unstable NiMn phase during the ageing process based on the observations of the alloy during the entire aging process, suggesting that the Ni$_3$Mn phase is more stable than the NiMn phase. More precipitates appear as the ageing time increases. In addition, the size of the precipitates do not change significantly after ageing for a long time, which may be the reason why the specimen still exhibits a high strength after ageing for a long time.
The hardness of an alloy is considered to be the combined effect of the superposition of multiple strengthening mechanisms, i.e., the solid solution strengthening, grain boundary strengthening, precipitation strengthening and work hardening mechanisms [16–18]. Grain boundary strengthening can be ignored since grain boundary strengthening contributes little during the ageing process. Solid solution strengthening can be expressed [19] as:

\[ \Delta \sigma_{ss} = M G_0 \left[ \left( \frac{\alpha_1 - \alpha_0}{a_0} \right)^2 + \left( \frac{2 G_1 - G_0}{G_1 + G_0} \right)^2 \right] c^2 \]  

(1)

where \( M \) is the Taylor factor, \( G_0 \) is the shear modulus of the copper matrix, \( a_0 \) is the lattice constant of the copper matrix, \( G_1 \) is the shear modulus of the solute, \( \alpha_1 \) is the lattice constant of the solute and \( c \) is the concentration of the solute in the alloys.

It can be seen from Equation (1) that solid solution strengthening is closely related to the content of solute. Consequently, the effect of solid solution strengthening will decrease if the solute content decreases. The large amount of precipitates that form during the ageing process continuously decreases the content of Ni and Mn in the matrix. Therefore, the contribution of solid solution strength should decrease during the ageing process.

The mechanism of work hardening can be expressed [20] as:

\[ \Delta \sigma_{\rho} = M \alpha G b \rho \frac{2}{3} \]  

(2)

where \( M \) is the Taylor factor, \( \alpha = 0.2 \) is the constant in copper alloys, \( G \) is the shear modulus of the matrix, \( b \) is the Burgers vector, and \( \rho \) is the dislocation density.

The effect of work hardening is closely related to the dislocation density according to Equation (2). The contribution of work hardening will decrease with decreasing dislocation density. The dislocation density decreases during the ageing process because of the recovery and recrystallization that occurs in the matrix and the generation of discontinuous precipitates. Therefore, the contribution of work hardening decreases during the ageing process.

Therefore, it is suggested that the increase in hardness is mainly contributed by precipitation strengthening, since the hardness of the alloy further increases during the ageing process and the other strengthening effects decrease according to the discussions above.

Precipitation strengthening is mainly caused by hindering the movement of dislocations, the mechanisms of which are mainly divided into two types, namely, the Orowan mechanism (bypassing) and the cutting mechanism [21,22]. The sizes of NiMn phases are approximately 5–10 nm. Therefore, the cutting mechanism is probably considered the main strengthening mechanism of the small NiMn precipitates, since the sizes of NiMn phases are quite small (approximately 5–10 nm). Compared with
the NiMn phase, the Ni$_3$Mn phase is larger in size; thus, the sample will be strengthened by the Ni$_3$Mn precipitates in the form of dislocation bypassing.

4.2. **Effect of Rolling Process on Microstructure and Properties of Cu-20Ni-20Mn Alloys**

The mechanical properties and ageing precipitation process of the alloy will be affected by cold deformation, which has been observed in the previous results. The hardness of the rolled specimen is greatly improved compared with that of the solid solution sample. The microstructure of the rolled specimen is found to transform from a columnar crystal structure to a slender fibrous structure, which is beneficial for obtaining many fine grains. In addition, the dislocation density is also greatly increased by the rolling process. Grain boundary strengthening and work hardening are the main reasons for the increase in hardness. Therefore, the precipitation process of the specimen during ageing is also influenced by cold deformation.

The precipitation rate during the ageing process of the sample increases through the effect of cold deformation. The time required to reach the peak strength during the ageing process of the rolled samples is greatly reduced compared with that of the solid solution samples. The incubation period is not found during the ageing process of the rolled sample, while it can be found during the ageing process of the solid solution sample. In addition, the hardness of the rolled sample at the peak ageing state is higher than that of the solid solution state. These changes in hardness are caused by the evolution of the microstructure. According to the results above, the deformation storage energy of the sample increases after the rolling process, which is beneficial for increasing the precipitation rate. In addition, many fine grains and a high dislocation density are obtained by the rolling process, which is beneficial for obtaining more precipitated phases because the number of nucleation sites increase during the ageing process. According to the discussion above, the mechanical properties of the alloy are improved by decreasing the grain size of the sample and increasing the dislocation density by rolling deformation. In addition, the precipitation rate of the rolled sample is greatly increased during the ageing process; thus, more precipitates can be obtained during the ageing process of the rolled sample.

5. **Conclusions**

A new compact preparation process involving heating-cooling combined mould (HCCM) continuous casting, a solution treatment at 800 °C, rolling at room temperature, and a final ageing step at 450 °C was developed to produce a Cu-20Ni-20Mn alloy. Compared with the traditional preparation process, the proposed process eliminated two hot deformation processes, namely, hot forging and hot rolling, which greatly improved the production efficiency and reduced production costs. The alloy fabricated by this new process was found to have excellent mechanical properties, and the hardness of the samples reached 476 HV. The following conclusions can be drawn from the results presented in this paper:

1. Two types of precipitates, i.e., NiMn and Ni$_3$Mn, could be observed in the aged sample. The structure of the Ni$_3$Mn phase is face-centred cubic (L1$_2$ structure), with the orientation relationship between the copper matrix and ordered Ni$_3$Mn phase (L1$_2$ phase) is (200)$_{\text{Cu}}$/[(010)$_{\text{Ni3Mn}}$] and [011]$_{\text{Cu}}$/[011]$_{\text{Ni3Mn}}$.

2. Precipitation strengthening is the main strengthening mechanism of the ageing process. The cutting mechanism is the main strengthening mechanism of the small NiMn precipitates. The Ni$_3$Mn phase is larger in size, suggesting that the sample will be strengthened by the Ni$_3$Mn precipitates in the form of dislocation bypassing.

3. The ageing precipitation process and mechanical properties of the alloys are affected by the rolling process. The precipitation rate of the rolled sample is greatly increased during the ageing process. The time required to reach the peak ageing during the ageing process of rolled samples is greatly reduced compared with solid solution samples. The incubation period is not found during the
aging process of the rolled sample, which could be found during the ageing process of the solid solution sample. In addition, more precipitates could be obtained during the ageing process of the rolled sample, which led to the hardness of the rolled sample in the peak ageing state being higher than that of the solid solution state.

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