Geological characteristics of shale in the Silurian (Sichuan) basin and chemical reaction of its reservoir under the action of fracturing fluid

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Abstract
The exploitation and exploration of shale gas is of great values to solve the energy problem. Taking the shale from Silurian Longmaxi formation in Sichuan Basin as an example, its geological characteristics and reservoir chemical reaction under the action of fracturing fluid were analyzed, and moreover, a series of data determination and fracturing fluid chemical reaction experiments were carried out. The results showed that the average total organic carbon value of shale in the study area was 4.79%, the kerogen type was type I, the ratio of aliphatic structure to aromatic structure was smaller than 1, the average R0 value was 2.15%, the content of clay mineral in the mineral composition was high, the average porosity was 3.16%, and the average permeability was 0.036 × 103 μm2, which was conducive to shale gas generation; under the action of fracturing fluid, the sulfate mineral in the shale dissolved, clay mineral expanded, and the pore volume and specific surface area reduced. The results verify that the research area has shale gas mining value, and this study makes some contributions to further study the optimization of fracturing fluid and improve exploitation technology.

Keywords Shale · Fracturing fluid · Sichuan basin · China

Introduction
With the development of economy, the problem of energy shortage has become more important, and the demand for energy is increasing; hence, the development of new energy has attracted more and more attentions, for example, oil–gas exploration (Al-Fatlawi et al. 2017a, b). The world natural gas market is developing rapidly, and its proportion in the world energy is expected to increase to 27% in 2020 (Dmitrievsky 2005). However, most oil and gas reservoirs are rapidly consuming (Al-Fatlawi et al. 2017a, b), and natural gas is a non-renewable resource (Martin-Gil et al. 2019). Shale gas, a kind of methane based, very important clean energy, is a potential energy which develops in a sustainable way (Shar et al. 2018) and has long development life and rich resources, which is an ideal substitute for natural gas. It is found that China is rich in shale gas resources, and there are shale gas accumulation conditions in Junggar Basin (Pang et al. 2018), Ordos Basin (Zhao et al. 2016), Sichuan Basin (Zhang et al. 2015), etc., which has a very broad development prospect. Lin et al. (2016) analyzed the shale of Qiongzhusi, Wufeng and Longmaxi formation of Lower Paleozoic in the south of Sichuan and the north of Yunnan and found that their average quartz content was more than 27% and the relative brittleness index was also very high, i.e., they had good fracturing property. In the process of shale gas exploitation, due to the low permeability of the reservoir, it is necessary to form artificial fractures through large-scale fracturing (Maity 2015; Kahrilas et al. 2016) to obtain industrial gas flow. Various fracturing technologies have been studied extensively (Al-Fatlawi et al. 2019; Dai et al. 2019), but the fracturing fluid used in the production will often remain in the reservoir (Zhang et al. 2017), affecting the fracturing effect. Eveline et al. (2016) pointed out that the clay expansion pressure increased after the fracturing fluid intruded into the matrix under the permeability, resulting in the decrease in shale permeability and the formation of failure areas. Luek et al. (2018) found that there were a large number of halogenated organic molecular particles in the backflow fluid of fracturing fluid and
the iodine ion in the backflow fluid continued to increase with the extension of time. This study analyzed the shale in Sichuan Basin from aspects of shale geological characteristics and the chemical reaction between fracturing fluid and reservoir. The geological characteristics of the shale in the research area were studied through X-ray diffractometer, and the chemical reactions between the shale and fracturing fluid were studied to obtain some information such as the organic geochemical characteristics and mineral composition of the shale. The changes of mineral composition and pore characteristics under the action of fracturing fluid were observed. This study aims to provide some theoretical supports for reasonable exploitation of shale gas and further improvement of exploitation level.

Overview of the study area

The limestone of Paleozoic and pre-Paleozoic is widely distributed in Sichuan Basin, and there is a large area of Mesozoic purplish red sandstone and mudstone and six sets of mature regional organic rich shale (Dong et al. 2015). The main research area of this study was Longmaxi formation of Lower Silurian system, whose main lithology was shale mixed with sandstone and limestone. The shale in that area was well exposed, which could be cored in large scale. The geographical location and lithologic profile of the study area is shown in Fig. 1.

Experimental method

Geological characteristics survey

Sichuan Basin is a part of the Yangtze platform. In the late Permian, magma erupted in the western part of the basin, forming a basalt eruption area (Long et al. 2011). The main research area of this study was Longmaxi formation of Lower Silurian system, whose main lithology was shale mixed with sandstone and limestone. The shale in that area was well exposed, which could be cored in large scale. The geographical location and lithologic profile of the study area is shown in Fig. 1.

Chemical reaction experiment

Under the long-term infiltration of fracturing fluid, chemical reaction will occur in shale, which can change the mineral and pore structure of shale and affect the exploitation of shale gas. Six samples (Fig. 2) were taken from the study area, and samples which came from the same layer and were adjacent to the six samples were taken for comparative analysis. The sample surface was cleaned using distilled water and then they were put into the oven for drying. The formula of the fracturing fluid used in the experiment was 0.2% domestic resistance reducer (for reducing pressure loss), 0.15% cleanup additive, 0.05% demulsifier and 1% antiswelling agent (for preventing swelling and migration.
The experiment was performed in a high temperature and pressure stainless batch tank reactor which has polytetrafluoroethylene equipment liner. There was 90 mL of solvent, and it could bear a maximum pressure of 40 MPa. The sample was put at the bottom of the reactor. Then, 40 mL of fracturing fluid was taken by a pipette and added to the reactor. The reaction temperature was 90 °C. The pressure in the reactor was controlled through constant flow pump. The reaction period was 24 h, 48 h and 72 h. The sample and solution were taken out at the end of the experiment.

**Research results**

**Geological characteristics of shale**

**Organic geochemical characteristics**

The higher the content of total organic carbon (TOC), the greater the adsorption capacity of shale gas. It was found that the content of TOC of shale in the study area was relatively high (Table 2), between 1.06 and 6.64%, and the average content of TOC was 4.79%. Kerogen can be divided into type I, type II and type III according to the elemental composition. It was found through analysis that kerogen in the study area was type I, which was more likely to produce hydrocarbon gas; the ratio of aliphatic structure to aromatic structure in kerogen was smaller than 1, indicating strong humic feature, and the \( R_O \) value was between 1.51 and 2.87% (average 2.15%), which was conducive to the formation of shale gas.

**Mineral composition of reservoir**

The results of XRD analysis on 100 samples from Longmaxi formation are shown in Table 3. It is seen from Table 3 that the mineral composition of different samples was different; some samples were mainly composed of clay minerals (e.g., X-2 and X-98), some samples were mainly composed of quartz (e.g., X-10), and some samples were mainly composed of carbonate minerals (e.g.,...
X-1 and X-4); overall, the content of clay mineral in the shale was relatively high, between 6.7 and 91.2%, with an average value of 49.6%, the content of quartz was between 0.3 and 72.6%, with an average value of 23.6%, and the content of carbonate mineral was between 0.2 and 81.7%, with an average value of 14.2%.

### Pore characteristics

Original information obtained by nuclear magnetic resonance method cannot be directly used, and useful information can only be obtained by T2 spectrum through inversion. T2 distribution map can reflect parameters of reservoir such as porosity and permeability, providing important information for reservoir evaluation. Four samples were taken as examples, and their T2 distribution maps are shown in Fig. 4. It is seen from Fig. 4 that T2 spectra showed bimodal distribution, and the second peak was at 10 ms, indicating that there were some large pores in the reservoir besides small pores; after centrifugation, T2 spectra showed unimodal distribution, indicating that the pores were mostly small pores with few movable fluid. The porosity of the samples was between 1 and 5% (Table 4) (average 3.16%), the permeability was between 0.001 × 10⁻¹³ and 0.100 × 10⁻¹³ μm² (average 0.036 × 10⁻¹³ μm²), showing good reservoir forming conditions.

### Table 2 Organic geochemical parameters of samples

| Sample number | TOC content/% | Kerogen type | Ratio of aliphatic structure to aromatic structure of kerogen | Rço/% |
|---------------|---------------|--------------|-------------------------------------------------------------|-------|
| Y-1           | 6.52          | I            | 0.51                                                        | 1.75  |
| Y-2           | 1.36          | I            | 0.64                                                        | 2.12  |
| Y-3           | 5.48          | I            | 0.53                                                        | 1.51  |
| Y-4           | 4.36          | I            | 0.72                                                        | 1.67  |
| Y-5           | 6.62          | I            | 0.89                                                        | 1.78  |
| Y-6           | 6.36          | I            | 0.76                                                        | 2.45  |
| Y-7           | 6.64          | I            | 0.84                                                        | 2.36  |
| Y-8           | 5.18          | I            | 0.56                                                        | 2.58  |
| Y-9           | 6.49          | I            | 0.74                                                        | 2.87  |
| Y-10          | 2.36          | I            | 0.64                                                        | 2.67  |
| Y-11          | 4.52          | I            | 0.56                                                        | 2.32  |
| Y-12          | 4.56          | I            | 0.62                                                        | 2.15  |
| Y-13          | 5.49          | I            | 0.76                                                        | 2.22  |
| Y-14          | 6.56          | I            | 0.71                                                        | 1.78  |
| Y-15          | 3.68          | I            | 0.59                                                        | 1.68  |
| Y-16          | 2.68          | I            | 0.97                                                        | 1.57  |
| Y-17          | 3.68          | I            | 0.84                                                        | 2.07  |
| Y-18          | 5.69          | I            | 0.55                                                        | 2.17  |
| Y-19          | 6.56          | I            | 0.49                                                        | 2.55  |
| Y-20          | 1.06          | I            | 0.68                                                        | 2.63  |

### Table 3 The mineral composition of samples

| Sample number | Clay mineral content/% | Quartz content/% | Carbonate mineral content/% |
|---------------|------------------------|------------------|-----------------------------|
| X-1           | 6.8                    | 26.7             | 66.5                        |
| X-2           | 86.3                   | 1.3              | 12.4                        |
| X-3           | 71.2                   | 5.6              | 23.2                        |
| X-4           | 17.2                   | 8.9              | 73.9                        |
| X-5           | 56.3                   | 11.5             | 32.2                        |
| X-6           | 37.6                   | 54.3             | 8.1                         |
| X-7           | 48.3                   | 31.2             | 20.5                        |
| X-8           | 36.7                   | 27.8             | 35.5                        |
| X-9           | 75.1                   | 12.6             | 12.3                        |
| X-10          | 12.5                   | 64.8             | 22.7                        |
| X-11          | 56.8                   | 14.6             | 28.6                        |
| X-12          | 64.7                   | 6.9              | 28.4                        |
| X-13          | 36.8                   | 42.8             | 20.4                        |
| ...           | ...                    | ...              | ...                         |
| X-95          | 71.5                   | 2.6              | 25.9                        |
| X-96          | 42.6                   | 7.9              | 49.5                        |
| X-97          | 51.2                   | 25.6             | 23.2                        |
| X-98          | 89.6                   | 5.6              | 4.8                         |
| X-99          | 11.5                   | 45.8             | 42.7                        |
| X-100         | 21.9                   | 24.7             | 53.4                        |
The comparison of the mineral composition of the original shale and the shale after the reaction with the fracturing fluid is shown in Table 5. It was found that the content of calcite and dolomite in the shale decreased significantly after the reaction with the fracturing fluid, and the changes of the other components were small.

Due to the dissolution of sulfate minerals, i.e., calcite and dolomite, in the shale, the pH value of the drainage solution changed. The pH value of the discharged fluid in different time periods was determined. The pH value of the original fracturing fluid was 4.7, i.e., the fluid was acid. The pH value of the discharged fluid was 6.5 after 24 h and 6.7 after 48 h, i.e., the fluid was neutral. After 72 h, it was 6.4.

In the fracturing fluid, a large amount of H⁺ was released from the resistance reducer by hydrolysis, which reacted with sulfate minerals. The process is as follows:

\[
\text{CaCO}_3 + 2\text{H}^+ = \text{Ca}^{2+} + \text{CO}_2 + \text{H}_2\text{O}
\]

\[
\text{CaMg(CO}_3\text{)}_2 + 4\text{H}^+ = \text{Ca}^{2+} + \text{Mg}^{2+} + 2\text{CO}_2 + 2\text{H}_2\text{O}
\]

In addition to the change of mineral composition, the pore characteristics of the shale changed under the influence of the fracturing fluid. As shown in Table 6, after chemical reaction, the pore volume and specific surface area of the shale decreased significantly, and the reason for such a phenomenon was that clay minerals expanded after reacting with the fracturing fluid, blocking the original pores in the shale.

### Table 4 The porosity distribution of samples

| Porosity (%) | 1–2% | 2–3% | 3–4% | 4–5% |
|--------------|------|------|------|------|
| Number of samples/n | 37 | 18 | 28 | 17 |

### Chemical reaction results of shale and fracturing fluid

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### Table 5 Changes of shale mineral composition (A represents the original shale, A* represents the shale after reaction with fracturing fluid, the same below)

| Sample number | Clay minerals/% | Minerals/% | Illite/smectite formation | Chlorite | Calcite | Dolomite | Pyrite | Plagioclase |
|---------------|----------------|-----------|--------------------------|----------|---------|----------|--------|------------|
|               | Total content of clay | Quartz | Calcite | Dolomite | Pyrite | Plagioclase |
| A             | 15 | 97 | 3 | 36 | 35 | 8 | 3 | 3 |
| A*            | 18 | 96 | 4 | 37 | 31 | 6 | 3 | 5 |
| B             | 16 | 96 | 4 | 36 | 36 | 7 | 2 | 3 |
| B*            | 19 | 95 | 5 | 37 | 32 | 5 | 2 | 5 |
| C             | 16 | 97 | 3 | 35 | 35 | 7 | 3 | 4 |
| C*            | 19 | 95 | 5 | 36 | 31 | 5 | 3 | 6 |
Discussion

Sichuan Basin is an important natural gas production area in China and a shale gas exploration test base. It has formed a good industrial production capacity and initially formed a mining and management technology in line with the geological characteristics of shale gas in the basin, with excellent drilling and fracturing reformation technologies. It has established orderly mining plans, standardized the production process, and made a great contribution to the local economic development.

This paper mainly analyzed the shale of Longmaxi formation, including the analysis of geological characteristics and chemical reaction of shale reservoir. The analysis results of geological characteristics of the shale demonstrated that the TOC content in the study area was relatively high, the main kerogen was type I, the maturity of organic matter was also relatively good, most of which were clay minerals, and the development of pores was good, providing a good condition for shale gas generation. In addition, the gentle terrain and absence of large-scale fault zones provide good preservation conditions, which is very helpful for shale gas development and exploration. The chemical reaction between shale reservoir and fracturing fluid showed that the retention of fracturing fluid in shale changed the mineral composition. In the process of reaction, water molecule entered minerals along the crack to form water molecule layer to absorb water, leading to uneven expansion, disintegration and dispersion. The dissolution of sulfate minerals reduced the porosity, which was not conducive to exploitation. In the mineral composition of shale, illite and montmorillonite expanded by absorbing water, destroying the pores and leading to reduced pore volume and specific surface area and dolomite and calcite dissolved, destroying the reservoir structure, which had a great influence on shale gas exploitation. Therefore, in the process of fracturing, the composition of chemical components of the fracturing fluid should be paid attention to improve the efficiency of fracturing mining. At present, in shale gas exploitation in China and abroad, the flowback rate of fracturing fluid is relatively low (Yan et al. 2015), and about 10–30% fracturing fluid returns to the ground (Tikhomirova et al. 2011). In addition to affecting the shale gas exploitation effect, the remaining fracturing fluid may also contain a large number of salts (Cl, Br) which will promote the dissolution of carbonate minerals (Joewong et al. 2015) and metal pollutant (Countess et al. 2014) and heavy metal pollutant (Dustin et al. 2018) which will change the permeability of rock (Harrison et al. 2017). Fletcher (2012) pointed out that the overflow of fracturing fluid brought large pollution risk to the underground water resource of Pennsylvania. Therefore, fracturing chemical products have begun the transition to being the least harmful to the environment and human health (Hurley et al. 2016). Further improvement of fracturing technology is the key point in shale gas production (Huang 2019).

Great achievements have been made in shale gas development in Sichuan Basin, but there are still some challenges. For example, in addition to the developed mature shale gas areas, the distribution of shale gas in other areas of Sichuan Basin is not clear, and there are still some technical restrictions on the development of deep shale gas resources. But in general, Sichuan Basin is rich in shale gas resources, showing a good development prospect, and the large-scale exploration and mining can be carried out in Sichuan Basin, but the flowback of fracturing fluid should be paid attention to.

Conclusion

Taking the shale of Longmaxi formation as an example, this paper analyzed its geological characteristics and chemical reactions. It was found that:

1. The TOC content of the shale in the study area was between 1.06 and 6.64%, the kerogen type was type I, and R_o value was between 1.51 and 2.87%, which was conducive to shale gas enrichment;
2. The shale in the study area had high clay mineral content and strong adsorption capacity for gas;
3. The shale porosity was between 1 and 5%, and the permeability was between 0.001 × 10^{-3} and 0.100 × 10^{-3} μm^{2}, which provided good reservoir forming conditions;
4. When the shale reacted with the fracturing fluid, sulfate minerals dissolved and pore volume and specific surface area decreased.

Compliance with ethical standards

Conflict of interest The authors declare that they have no conflict of interest.

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Table 6 Pore characteristics of shale

| Sample number | Pore volume (cm^3/g) | Specific surface area (m^2/g) |
|---------------|---------------------|------------------------------|
| A             | 0.01346             | 13.54                        |
| A*            | 0.00864             | 8.24                         |
| B             | 0.01354             | 12.78                        |
| B*            | 0.00798             | 8.64                         |
| C             | 0.01365             | 13.62                        |
| C*            | 0.00852             | 8.53                         |
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