Mangiferin, a Natural Xanthone, Protects Murine Liver in Pb(II) Induced Hepatic Damage and Cell Death via MAP Kinase, NF-κB and Mitochondria Dependent Pathways

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Abstract

One of the most well-known naturally occurring environmental heavy metals, lead (Pb) has been reported to cause liver injury and cellular apoptosis by disturbing the prooxidant-antioxidant balance via oxidative stress. Several studies, on the other hand, reported that mangiferin, a naturally occurring xanthone, has been used for a broad range of therapeutic purposes. In the present study, we, therefore, investigated the molecular mechanisms of the protective action of mangiferin against lead-induced hepatic pathophysiology. Lead (Pb(II)) in the form of Pb(NO₃)₂ (at a dose of 5 mg/kg body weight, 6 days, orally) induced oxidative stress, hepatic dysfunction and cell death in murine liver. Post treatment of mangiferin at a dose of 100 mg/kg body weight (6 days, orally), on the other hand, diminished the formation of reactive oxygen species (ROS) and reduced the levels of serum marker enzymes [alanine aminotransferase (ALT) and alkaline phosphatase (ALP)]. Mangiferin also reduced Pb(II) induced alterations in antioxidant machineries, restored the mitochondrial membrane potential as well as mutual regulation of Bcl-2/Bax. Furthermore, mangiferin inhibited Pb(II)-induced activation of mitogen-activated protein kinases (MAPKs) (phospho-ERK 1/2, phosphor-JNK phospho- p38), nuclear translocation of NF-κB and apoptotic cell death as was evidenced by DNA fragmentation, FACS analysis and histological assessment. In vitro studies using hepatocytes as the working model also showed the protective effect of mangiferin in Pb(III) induced cytotoxicity. All these beneficial effects of mangiferin contributes to the considerable reduction of apoptotic hepatic cell death induced by Pb(II). Overall results demonstrate that mangiferin exhibit both antioxidative and antiapoptotic properties and protects the organ in Pb(II) induced hepatic dysfunction.

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Introduction

Lead, a naturally occurring and one of the most well-known heavy metals in the environment, has been reported to cause potential danger to human health [1,2]. It is a universal toxic metal that affects several organs (like liver, kidneys, etc.) and the hematopoetic, central nervous, endocrine as well as reproductive systems [3] of our body. Humans are exposed to lead generally via food, water, inhalation of lead-contaminated dust particles or aerosols in the working place [3,4]. It is also used in crystal and ceramic containers that leach into water and food in deteriorating household paints, in some traditional medicine and cosmetics [4,5]. Like other naturally occurring heavy metals-arsenic, mercury, and cadmium-lead also damages cellular substance and changes cellular genetics. The mechanism of lead toxicity, in general, involves oxidative damage that affects cell membrane and activates factors susceptible to transcription [6,7,8]. Several studies on lead nitrate [Pb(NO₃)₂] exposure showed that it produces reactive oxygen species (ROS) and disturbs the prooxidant-antioxidant balance. Usually Pb(II) binds to sulfhydryl (-SH) groups of biomolecules, disrupts structural protein synthesis, changes calcium homeostasis, and lowers the level of available sulfhydryl antioxidant, reduced glutathione (GSH) stores in the body [9]. But the mechanism of Pb-induced hepatotoxicity is not very clear. Results from some recent studies intensely propose that oxidative stress [10,11,12] and cellular apoptosis [13] are the main causes for hepatic pathophysiology. Lead-induced hepatic damage typically caused lipid peroxidation via the production of ROS [14]. It decreases the activities of several antioxidant enzymes like catalase (CAT), superoxide dismutase (SOD), glutathione peroxidase (GPx), glutathione reductase (GR), etc. [15].

Medicinal plants are of enormous importance to the health of human beings and the therapeutic value of these plants is mainly due to the presence of some chemically active materials that generate a specific physiological action in our body. The most important bioactive ingredients of plants are: flavonoids, tannins, alkaloids and phenolic compounds [16]. A number of medicinal plants and herbs such as Silybum marianum [17], Terminalia arjuna [18–21], Cajanus indicus [22–29], Phyllanthus niruri [30–39], Pithecellobium dulce [40,41], etc. in India and other parts of the World are natural sources of antioxidants that act as the first line of defense against free radical damage and are considered to be important in maintaining optimum health and happiness. These are used for the treatment of hepatic, renal and other organ disorders. Polyphenolic compounds and flavonoids are abundant in fruits, vegetables, tea and wine. These compounds are usually
recognized to have powerful antioxidant properties [42]. In connection with these plants and herbs, another important medicinal plant available throughout the world is *Mangifera indica* L. It belongs to the family Anacardiaceae and is the source of many natural xanthenes, polyphenols etc. We have isolated and characterized mangiferin, [2-C-[β-Dgluco-β-D-glucopyranosyl-1,3,6,7-tetrahydroxyxanthenone; C_{19}H_{18}O_{11}; Mw, 422.35; melting point, anhydrous 271°C [43], a natural C-glucoside xanthone [44] from its bark (*Mangifera indica* L.). A number of studies reported that mangiferin has a broad range of therapeutic uses. It possesses antioxidant [45,46,47], anti-diarrhoea [48], dyslipidemic [49], antidiabetic [50], anti-arrhythmic [51], antibacterial [52], anti-HIV [53] and anticancer [54] activities. Besides, it is also used as an analgesic, immunomodulatory [55] and immunostimulatory [56], agents.

The aim of the present study was to investigate the mechanisms underlying the protective action of mangiferin in lead(II)-induced hepatic pathophysiology using both in vivo and in vitro working models. First of all, radical scavenging activity of mangiferin was determined by DPPH radical, superoxide radical, nitric oxide radical and hydroxyl radical scavenging assays. Lead-nitrate-induced liver injury and oxidant-antioxidant status was assessed by measuring liver specific serum marker enzymes (ALT and ALP) leakages; lipid peroxidation, protein carbonylation; levels of cellular metabolites (GSH and GSSG) and activities of antioxidant enzymes (CAT, SOD, GST, GPX, GR etc). Underlying cell signaling mechanism was determined by investigating the anti-apoptotic Bcl-2 and pro-apoptotic Bax proteins, cytosolic cytochrome C, caspase 3 as well as caspase 9 protein levels. Role of mitogen-activated protein kinase (MAPKs) and NF-κB under this pathophysiological state and the protective action of mangiferin was also investigated in this study. The nature of cell death by Pb(NO₃)₂ induced hepatotoxicity and its protection by mangiferin has been investigated by DNA fragmentation and FACS analysis. The results of the present study are expected to provide a clear picture about the role of mangiferin in lead-nitrate-induced liver injury, and may shed light on an achievable solution to the serious liver problems caused by lead exposure.

**Materials and Methods**

**Chemicals**

Anti-JNK, anti-p38, anti-ERK1/2, phosphorylated ERK1/2, anti-Bcl-2, anti-Bad, anti-caspase 3, anticaspase 9, anti-NF-κB, anti Apaf1 and anti cytochrome C antibodies were purchased from Sigma-Aldrich Chemical Company (St. Louis, USA). Kits for ALP and ALT measurements were purchased from Span diagnostic Ltd., India. Pb(NO₃)₂ and other essential reagents of analytical grade were bought from Sisco Research Laboratory, India.

**Animals**

Healthy adult male albino mice (30) of Swiss strain, weighing between 20 and 25 g were purchased from M/S Ghosh enterprise, Kolkata, India. The animals were adjusted under laboratory conditions for a fortnight before starting experiments. The animals were maintained in a standard diet and water ad libitum. They were housed in polypropylene cages and exposed to 10–12 h of daylight under standard conditions of temperature (25°C) and humidity (30%). All the experiments with animals were carried out according to the guidelines of the institutional animal ethical committee and full details of the study was approved by the CPCSEA, Ministry of Environment & Forests, New Delhi, India (the permit number is: 95/99/CPCSEA).

**Extraction and isolation of mangiferin**

Mangiferin was extracted in our laboratory following the method as described by Ghosh et al (2012) [57]. Briefly, crudely powdered bark of *Mangifera indica* was extracted exhaustively with ethanol (95%) in soxhlet apparatus for 56 h. The combined alcohol extracts were concentrated under reduced pressure when a yellow amorphous powder was obtained. The dried alcoholic extract was adsorbed on silica gel (60–120 mesh) and chromatographed over silica gel column packed in petroleum ether (60–80°C). The column was eluted with chloroform: methanol (1:1) which gave mangiferin as a pale yellow amorphous powder. This upon crystallization from ethanol produced pale yellow needle shaped mangiferin crystals. Homogeneity of preparation was checked by the HRMS (ESI) analysis, HPLC and NMR (1H, 13C) spectroscopy.

**Determination of radical scavenging activity of mangiferin in cell-free system**

**DPPH radical scavenging activity.** The DPPH radical scavenging activity of mangiferin has been measured by the method of Blois [58]. Two ml of DPPH solutions (125 μM) in methanol and 2 ml of tested samples with different concentrations (0.5, 1, 5, 10, 20, 30, 40, 50 and 60 µg/mL) of mangiferin were mixed in the tubes. The solution was incubated at 37°C for 30 minutes in dark. The decrease in absorbance at 517 nm was measured against methanol blank using a spectrophotometer. Vitamin C was used as a positive control.

**Superoxide radical scavenging activity.** The superoxide radical scavenging activity of mangiferin was measured by the method as described elsewhere [59]. In brief, different concentrations of mangiferin were mixed with 0.1 M phosphate buffer pH 7.4, 150 μM nitroblue tetrazolium (NBT), 60 μM phenazine methosulphate (PMT) and 48 μM NADH. The mixture was incubated 10 minutes at 25°C in the dark and the absorbance was read at 560 nm. Results were expressed as percentage inhibition of the superoxide radicals. Quercetin was used as a standard for the study.

**Nitric oxide radical scavenging activity.** The nitric oxide radical scavenging activity of mangiferin has been measured following the methods as described elsewhere [60,61]. Sodium nitroprusside (SNP) in aqueous solution at physiological pH spontaneously generates nitric oxide. Under aerobic conditions, nitric oxide reacts with oxygen to produce stable products nitrate and nitrite ions that can be estimated by using of Griess reagent. Briefly, the reaction mixture in phosphate buffered saline (pH 7.4) containing 10 mM SNP and various doses (0-60 µg/mL) of the test solution in a final volume of 3 ml were incubated for 150 min at 25°C. One ml sulfanilamide (0.33% in 20% glacial acetic acid) was added to 0.5 mL of the incubated solution and allowed to stand for 5 min. One mL of naphthylethendiamine dihydrochlorode (NED) (0.1% w/v) was then added and the mixture was again incubated for 30 min at 25°C. The pink chromophore generated during diazotization of nitrite ions with sulphanilamide and subsequent coupling with NED was measured spectrophotometrically at 540 nm against a blank containing no test sample. Curcumin was used as a standard for this experiment.

**Hydroxyl radical scavenging activity.** The hydroxyl radical scavenging activity was measured by the method of Nash [62]. Hydroxyl radical concentration was determined in terms of formaldehyde generation from oxidation of dimethyl sulphoxide (DMSO). The amount of formaldehyde generated was measured in terms of diacetylthioldihydrolutidine formed by the action of acetyloacetone and ammonium acetate. A yellow colour is
developed and that is measured spectrophotometrically at 412 nm. Mannitol was used as a standard for this study.

### Determination of dose and time dependent activity of Pb(NO₃)₂ by ALT level

To set up the dose of Pb(NO₃)₂ needed for maximum damage in murine liver, mice were arbitrarily assigned into six groups each consisting of six mice and they were treated as follows. First group served as normal control (received only water as vehicle). Remaining five groups were treated with five different doses of Pb(NO₃)₂ orally (1 mg, 3 mg, 5 mg, 7 mg and 9 mg/kg body weight for 6 days).

To find out the time needed for Pb(NO₃)₂ induced maximum damage in murine liver, experiments were carried out with five groups of animals consisting six animals in each group. The first group received water as vehicle and served as normal control. Pb(NO₃)₂ was administered orally to other four groups at a dose of 5 mg/kg body weight for 2, 4, 6, and 8 days respectively.

Twenty-four hours after the final dose of Pb(NO₃)₂ intoxication, all mice were sacrificed. Serum activity of alanine transaminase (ALT) was determined photometrically using standard test kit (Span diagnostic Ltd., India).

### Determination of dose and time dependent activity of mangiferin by ALP level

For this study, mice were randomly distributed into eight groups each consisting of six animals. First two groups were served as normal control (received only water as vehicle) and toxin control (received Pb(NO₃)₂ at a dose of 5 mg/kg body weight for 6 days, orally) respectively. Remaining six groups of animals were treated with six different doses of mangiferin (25 mg, 50 mg, 75 mg, 100 mg, 125 mg and 150 mg/kg body weight for 6 days, orally) after Pb(NO₃)₂ intoxication (5 mg/kg body weight for 6 days, orally, once daily).

To determine the time dependent effects of mangiferin, mice were divided into nine groups each consisting of six animals. First two groups were served as normal control (received only water as vehicle) and toxin control (received Pb(NO₃)₂ at a dose of 5 mg/kg body weight for 6 days, orally) respectively. Other seven groups of animals were treated with mangiferin orally at a dose of 100 mg/kg body weight, once daily for 2, 3, 4, 5, 6, 7 and 8 days after Pb(NO₃)₂ intoxication (5 mg/kg body weight for 6 days, orally, once daily).

Twenty-four hours after the final dose of Pb(NO₃)₂ administration all mice were sacrificed. Serum activity of alkaline phosphatase (ALP) was determined spectrophotometrically using standard test kit (Span diagnostic Ltd., India).

### Experimental set-up for in vivo treatments

Experimental design needed for the present in vivo study has been summarised in Figure 1. The animals were divided into four groups each consisting of six mice and they were treated as follows.

#### Group 1: (Normal): Animals received only water as vehicle.

#### Group 2: (Only mangiferin treated Group): Mangiferin was administered orally at a dose of 100 mg/kg body weight for 6 days, once daily.

#### Group 3: (Toxin control): Animals received Pb(NO₃)₂ orally at a dose of 5 mg/kg body weight for 6 days, once daily.

#### Group 4: (Post-treated group): Animals were treated with a single dose of Pb(NO₃)₂ (orally, 5 mg/kg body weight, once daily) for 6 days followed by mangiferin treatment (orally, at a dose of 100 mg/kg body weight, once daily) for next 6 days.

### Preparation of liver tissue homogenates

The liver were minced, rinsed, and homogenized in a Dounce glass homogenizer in 10 mM HEPES-KOH/1 mM EGTA buffer (pH 7.5) containing 250 mM sucrose supplemented with protease and phosphatase inhibitors. The homogenates were spun down for...
Determination of protein content

The protein content of the experimental samples was measured by the method of Bradford [63] using crystalline BSA as standard.

Measurement of lipid peroxidation and protein carbonyl content

The lipid peroxidation in normal and experimental liver tissue homogenate (containing 1 mg of protein) in terms of malondialdehyde (MDA) formation was measured following the method of Esterbauer and Cheeseman [64]. The absorbance of thiobarbituric acid reactive substance (TBARS) formed as the end product, was measured at 532 nm and the concentration of the samples was calculated using the extinction coefficient of MDA as 1.56 \times 10^5 M^{-1} cm^{-1}. Like lipid peroxidation, protein carbonylation is also a marker of oxidative stress. In the present study, protein carbonyl contents were determined according to the methods Uchida and Stadtman [65]. The samples were treated with an equal volume of 0.1% (w/v) 2,4-DNPH in 2N HCl and incubated for 1 hour at room temperature. 20% TCA was then added for the precipitate formation which was collected by centrifugation. It was extracted three times with EtOH/EtOAc and dissolved in 8 M guanidine hydrochloride in 1 M tris solution containing 1 M EDTA. The absorbance was measured at 365 nm and the results were expressed as nmol of DNPH incorporated/mg protein using the molar extinction coefficient of aliphatic hydrazones as 22000 M^{-1} cm^{-1}.

Assay of cellular metabolites

Cellular GSH levels were determined by using Ellman’s reagent (DTNB; 5,5-dithiobis-2-nitrobenzoic acid) [66]. Oxidized glutathione GSSG contents in the experimental samples were determined following the method of Hissin and Hilf [67].

Table 1. Effect of Pb(II) and mangiferin on the activities of the lipid peroxidation and protein carbonylation in liver tissue.

| Parameters                        | Control          | Mangiferin       | Pb(II)           | Pb(II)+ Mangiferin |
|-----------------------------------|------------------|------------------|------------------|-------------------|
| MDA(nmol/mg protein)              | 9.78 ± 0.18      | 9.36 ± 0.18      | 22.36 ± 0.41a    | 11.76 ± 0.208b    |
| Protein Carbonylation(nmol/mg protein) | 14.16 ± 0.27    | 14.43 ± 0.299    | 44.63 ± 0.832a   | 21.83 ± 0.404b    |

Values are expressed as mean ± SEM, for 6 animals in each groups. "a" values differs significantly from normal control (P<0.05); "b" values differs significantly from Gal control (P<0.05).

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Isolation of primary mouse hepatocytes

Primary hepatocytes were isolated from mouse liver by the perfusion technique with collagenase type I at 37°C [72]. The preparations with cell viability (MTT assay) greater than 90% were used for the experiments. Cells were seeded onto culture plates precoated with collagen at a density of 2\times10^5 cells/well for 24 well plates and 1\times10^6 cells/well for 6 well plates. The cells were cultured and preserved in William’s medium E, supplemented with 0.3 μM insulin, 0.1 μM dexamethasone, 10% FBS and 1% penicillin/streptomycin at 37°C, 90% humidity and 5% CO_2. All experiments were carried out 24 h after cell attachment to allow the formation of monolayer cells.

Determination of dose-dependent activity of Pb(II) in hepatocytes

The cell viability assessment was performed to determine the optimum dose of Pb(NO_3)_2 for cytotoxicity. Five different sets of hepatocytes (1 mL cell suspension \sim 1\times10^6 cells in each) were incubated with Pb(NO_3)_2 at a dose of 30μg/mL at different times (1.0 hr, 1.5 hr, 2.0 hr, 2.5 hr and 3.0 hr). After the incubation periods, MTT assay was performed with these five sets following the method of Madesh M and Balasubramanian [73].

Determination of time-dependent activity of Pb(II) in hepatocytes

Time-dependent cytotoxicity of Pb(NO_3)_2 was measured by cell viability assessment. Five different sets of hepatocytes (1 mL cell suspension \sim 1\times10^6 cells in each) were incubated with Pb(NO_3)_2 at a dose of 30μg/mL at different times (1.0 hr, 1.5 hr, 2.0 hr, 2.5 hr and 3.0 hr). After the incubation periods, MTT assay was performed with these five sets following the method of Madesh M and Balasubramanian [73].

Determination of dose-dependent activity of mangiferin in hepatocytes

Cell viability assay has been performed to determine the optimum dose of mangiferin for cytoprotection. Six different doses of mangiferin (30, 40, 50, 60, 70 and 80μg/mL) were used against Pb(II)-induced (30μg/mL) cytotoxicity using six different sets of hepatocytes (1 mL cell suspension \sim 1\times10^6 cells in each) for each experiment. The cells were incubated for 2 hrs and MTT assay was carried out with these six sets according to the method of Madesh M and Balasubramanian [73].

Agarose gel electrophoresis for DNA fragmentation

The DNA fragmentation has been assayed by electrophoresing genomic DNA samples, isolated from normal as well as experimental liver, on agarose/EtBr gel by the procedure described by Sellins and Cohen [74].

Detection of mode of cell death (apoptosis/necrosis) by flow cytometry

For the detection of Pb(II) induced nature of cell death (apoptosis/necrosis) we have performed flow cytometric analysis using hepatocytes as the working model. Hepatocytes were rinsed with phosphate-buffered saline (PBS), centrifuged at 800 g for 6 min, resuspended in ice-cold 70% ethanol/PBS, centrifuged at 7000 g for 10 min and resuspended in 70% ethanol/PBS. The DNA was precipitated and resuspended in 300 μL of 1% Triton X-100/PBS (pH 7.4). Other reagents such as propidium iodide (PI) and RNAse A were added and the mixture was incubated for 5 min before flow cytometric analysis of DNA content.

Table 3. Effect of Pb(II) and mangiferin on the activities of the antioxidant enzymes in liver tissue.

| Name of the antioxidant enzymes | Control | Mangiferin | Pb(II) | Pb(II)+Mangiferin |
|--------------------------------|---------|------------|--------|-----------------|
| SOD (Unit/mg protein)         | 68.46±1.27 | 69.03±1.36 | 38.45±0.0702* | 62.23±1.27b   |
| CAT (μmol/min/mg protein)     | 270.51±6.94 | 269.83±6.79 | 165.62±3.39* | 245.48±4.906b  |
| GST (μmol/min/mg protein)     | 1.49±0.27  | 1.45±0.277 | 0.84±0.017*  | 1.39±0.265b   |
| Gpx (nmol/min/mg protein)     | 97.54±1.83 | 96.12±1.91 | 48.43±0.98*  | 93.37±1.84b   |
| Gpx. (nmol/min/mg protein)    | 91.37±1.808 | 90.23±1.587 | 50.49±1.01*  | 86.46±1.677b  |

Values are expressed as mean ± SEM, for 6 animals in each groups. "a" values differs significantly from normal control (P<0.05); "b" values differs significantly from Gal control (P<0.05).
and developed by the HRP substrate 3,3'-diaminobenzidine tetra hydrochloride (DAB) system (Bangalore genei, India).

800 g for a further 6 min, and resuspended in PBS. Cells were then incubated with propidium iodide (PI) and FITC-labelled Annexin V for 30 min at 37°C. Excess PI and Annexin V were then washed off; cells were fixed and then stained cells were analyzed by flow cytometry using FACS Calibur (Becton Dickinson, Mountain View, CA) equipped with 488 nm argon laser light source; 515 nm band pass filter for FITC-fluorescence and 623 nm band pass filter for PI-fluorescence using CellQuest software. A dot plot of PI-fluorescence (y-axis) versus FITC-fluorescence (x-axis) has been prepared. A dot plot of PI-fluorescence (y-axis) versus FITC- fluorescent (x-axis) has been prepared using six independent experiments for all sets of hepatocytes. Pictures of one set of these experiments have been presented in the manuscript.

Immunobloting analysis
Same amount of protein (50 μg) from each sample was resolved by 10% SDS-PAGE and transferred to PVDF membrane. Membranes were blocked at room temperature for 2 h in blocking buffer containing 5% non-fat dry milk to prevent non specific binding and then incubated with anti-p-38 (1:1,000 dilution), anti-ERK1/2 (1:1,000 dilution), anti-p-JNK (1:1000 dilution), anti Bcl-2 (1:1000 dilution), anti Bax (1:1000 dilution) and anti cytochrome C (1:1,000 dilution), anti-caspase 3, anti-caspase 9 (1:1000 dilution), anti Bcl-2 (1:1000 dilution), anti- Apaf1 (1:1000) primary antibodies separately and were analyzed by one-way ANOVA, with Student-Newman-Keuls post hoc tests. "a" indicates the significant difference between the normal control and Pb(II) induced groups and "b" indicates the significant difference between the Pb(II) induced and mangiferin treated groups. (P<0.05, P<0.05).

Figure 3. Effect of mangiferin on Pb(II) induced intracellular ROS production. Cont: normal control, Mang: treated with mangiferin, Pb(II): administered with Pb(II), Pb(II)+Mang: Mangiferin treated post to Pb(II) administration. Data are mean ± SEM, for 6 sets per group and were analyzed by one-way ANOVA, with Student-Newman-Keuls post hoc tests. "a" indicates the significant difference between the normal control and Pb(II) induced groups and “b” indicates the significant difference between the Pb(II) induced and mangiferin treated groups. (P<0.05, P<0.05).

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Isolation of mitochondria from liver tissue and determination of mitochondrial membrane potential (ΔΨm)
Mitochondrial membrane potential of the live tissue of the experimental mice was measured according to method of Chandrasekaran et al. (2009) [75]. Briefly, liver tissue was minced, homogenized in ice-cold isolation buffer (10 mL Tris-MOPS [0.1 M; pH 7.4], 20 mL sucrose [1 M], and 1 mL EGTA-Tris buffer [0.1 M; pH 7.4]). The homogenates were centrifuged at 600 g at 4°C for 10 min, supernatant was collected and once more centrifuged at 7,000 g at 4°C for 10 min. The supernatant was discarded, the pellet was rinsed one time with isolation buffer, and the centrifugation steps were repeated two times. After discarding the supernatant, 1 mL of isolation buffer was used to suspend the pellet. Mitochondrial membrane potential (ΔΨm) was estimated by using the fluorescent cationic probe rhodamine 123 [76]. The data were analyzed by Cell Quest software.

Histological studies
For histological studies, small portion of livers from the normal and experimental mice were fixed in 10% buffered formalin and were processed for paraffin sectioning. Sections of about 5 μm thickness were stained with haematoxylin and eosin to evaluate the pathophysiological changes under light microscope.

Statistical analysis
All the experimental values are expressed as mean ± SEM (n = 6). Significant differences between the groups were determined with SPSS 10.0 software (SPSS Inc., Chicago, IL, USA) for Windows using one-way analysis of variance (ANOVA) and the group means were compared by Duncan’s Multiple Range Test (DMRT). A difference was measured significant at the p<0.05 level.

Results
Free radical scavenging activities of mangiferin in cell free system
Effect on DPPH radical and superoxide radical scavenging activities. Figure 1A shows the DPPH and superoxide radicals scavenging activities of mangiferin in cell free systems. The maximum DPPH radical scavenging activity has been observed at a concentration of 20 μg/ml. It has also been observed that optimum concentration of mangiferin for super oxide radical scavenging activity was 30 μg/ml. In both the cases the radical scavenging activities of mangiferin were higher than that of standard compounds (Vit C and Quercetin).

Effect on nitric oxide and hydroxyl radical scavenging activities. Figure 1B reports the hydroxyl radical and nitric oxide scavenging activities of mangiferin in cell free systems. The optimum hydroxyl and nitric oxide radical scavenging activities of mangiferin have been observed at the concentrations of 30 μg/ml and 50 μg/ml respectively. In both the cases the radical scavenging activities of mangiferin were higher than that of standard compounds (Mannitol and Curcumin).

Dose and time dependent study of Pb(II) by ALT level
As the first step of determining the dose and time necessary for Pb(II) to induce maximum hepatic damage, we carried out the dose- and time- dependent assays by measuring serum ALT activity as the index of the liver damage. As evidenced from Figure 2A, Pb(II)-intoxication increased the ALT activity linearly up to a dose of 5 mg/kg body weight for 6 days. This dose and
time were, therefore, chosen as the optimum dose and time of Pb(II) throughout the study.

Dose and time dependent effect of mangiferin by ALP level

ALP assay was used to determine the optimum dose and time necessary for mangiferin for the protection of liver against Pb(II)-induced oxidative damage. Experimental results suggest that Pb(II) exposure increased the ALP level and that could be prevented by the post-treatment with mangiferin linearly up to a dose of 100 mg/kg body weight for 6 days (Figure 2B). This dose of mangiferin and time of treatment were used for the subsequent experiments.

Effects on lipid peroxidation and protein carbonylation

Lipid peroxidation and protein carbonylation are two important markers of oxidative stress. Lipid peroxidation has been measured by estimating the concentration of MDA (end product of lipid peroxidation). Pb(NO₃)₂ administration increased the levels of MDA and protein carbonylation in the liver tissue of the experimental mice (Table 1). Post-treatment with mangiferin after the Pb(II) exposure efficiently reduced these levels in the liver tissue.
Effect on cellular metabolites

Thiol based antioxidant system plays second line of cellular protection against reactive free radicals mediated oxidative damage in pathophysiological situation. Cellular metabolites like GSH and GSSG levels have been presented in Table 2. The level of GSH has been considerably decreased because of Pb(NO₃)₂ intoxication along with the increased level of GSSG. Post treatment with mangiferin after the Pb(II) exposure restored the levels of cellular metabolites close to normal demonstrating the protective nature of mangiferin for hepato-cellular protection due to Pb(II) exposure.

Effect on antioxidant enzymes

The effect of mangiferin on the activities of the antioxidant enzymes (CAT, SOD, GST, GPX and GR) in Pb(NO₃)₂ exposed liver have been shown in Table 3. The activities of these antioxidant enzymes in Pb(II) treated liver were significantly lower than that present in normal liver tissue. Post-treatment of mangiferin after Pb(NO₃)₂ administration could enhance the activities of these antioxidant enzymes.
Effect on intracellular ROS production

Disorders of the pro-oxidant and anti-oxidant equilibrium in favour of the former play an important role in organs pathophysiology. This state of affairs arises as a result of either the increased production of reactive oxygen species (ROS) or the decreased level of the antioxidant defense. Pb(II) administration generates excess ROS either directly or indirectly in organ pathophysiology. Figure 3 shows the intracellular ROS levels in the normal and experimental livers. In Pb(NO₃)₂ exposed liver, intracellular ROS level increased significantly. Treatment with mangiferin after the Pb(NO₃)₂ administration reduced that level compared to the Pb(II)-exposed liver.

Dose dependent effect of Pb(II) induced hepatocytes

Cell viability is an important indicator of finding the degree of cytotoxicity caused by any xenobiotics. Figure 4A shows the dose dependent effect of Pb(NO₃)₂ in murine hepatocytes. It has been observed that Pb(II) exposure caused a decrease in cell viability linearly up to a dose of 30 μg/ml when incubated for 2 hour. Effect of Pb(II) remained more or less unaffected beyond this concentration. Therefore this dose (30 μg/ml) of Pb(II) has been chosen for subsequent in vitro studies.

Time dependent activity of Pb(II) induced hepatocytes

Figure 4B demonstrates the results of time dependent effect of Pb(NO₃)₂ in mouse hepatocytes as obtained by MTT assay. From this study, we found that Pb(NO₃)₂ (30 μg/ml) caused a significant decrease in cell viability when incubation was carried out for 2 hours. Beyond this incubation time (2 hours), effect of Pb(II) remained practically unaltered. So this incubation time (2 hours) has been taken as optimum for subsequent in vitro experiments.

Dose dependent effect of mangiferin against Pb(II)-induced hepatocytes damage

Outcome of the dose dependent effects of mangiferin against Pb(II)-induced cytotoxicity in hepatocytes have been shown in the Figure 4C. Pb(II) exposure caused a decrease in cell viability at a dose of 30 μg/mL when incubation was achieved for 2 hours. To decide whether this loss could be prevented by mangiferin treatment, we carried out MTT assay. Simultaneous incubation of hepatocytes with mangiferin and Pb(II) linearly inhibits the decrease in cell viability up to a dose of 50 μg/mL for 2 hours and this effect of mangiferin stayed practically unaltered either side of this concentration. So this dose (50 μg/mL) was chosen for subsequent in vitro studies.

Effect on DNA damage

To demonstrate the nature of cell death, next we studied the DNA fragmentation analysis. Pb(NO₃)₂ administration caused a DNA ladder fragmentation, a hallmark of apoptosis (Figure 5). Post-treatment with mangiferin on the other hand could efficiently protect this DNA laddering.

Effect of mangiferin on Pb(II)-exposed apoptotic/necrotic death

To investigate whether the toxic effect of Pb(II) on hepatocytes viability and its protection by mangiferin involves the process of cell apoptosis and/or necrosis, hepatocytes of all groups were assessed by flow cytometric analysis. Flowcytometric data (Figure 6) revealed that Pb(II) intoxicated (compare with control) hepatocytes showed maximum Annexin V-FITC-binding (47.5% & 3.2%) representing that the nature of the Pb(II) exposed cell death was primarily apoptotic. In contrast, the number of apoptotic cells was extensively low (only 11.1%) in the cell populations exposed to concurrent incubation of mangiferin and Pb(II), demonstrating that mangiferin might protect hepatocytes from Pb(II)-induced apoptotic death.

Activation of MAP kinase

In signal transduction pathways, mitogen-activated protein kinases (MAPKs) are the upstream critical signaling proteins. To investigate the mechanism for Pb(NO₃)₂-induced liver injury, we, therefore, studied the role of these kinases (JNK, p38 MAP kinase, and ERK) in this pathophysiology. To assess the effect of Pb(II) exposure on the activation of MAPK subfamilies, the liver tissue homogenates were analyzed for both total and phosphorylated forms of JNK, p38 and ERK (1/2) by immunoblotting. From the left panel of Figure 7, it is clear that the alterations in the protein contents of the phosphorylated and total MAPKs induced by Pb(II) could be inhibited by the post-treatment with mangiferin. We also observed similar results for the in vitro studies using hepatocytes as the working model (Figure 7, right panel).

Involvement of NF-κB

The transcription factor NF-κB which controls the transcription of DNA, is delicately susceptible to cellular oxidative status and apoptosis in response to oxidative stress. So, we have investigated whether this transcription factor plays any role on Pb(II) induced hepatic pathophysiology and if there is a positive response, whether mangiferin could alter it. Pb(NO₃)₂ administration caused a substantial increase in the expression of NF-κB and decrease in the expression of IkBα when compared to the normal group together with phosphorylation of IκKα. On the other hand Pb(II)-induced increase expression of NF-κB, IκKα and decrease expression IkBα was found to be significantly altered by the post-treatment with mangiferin (Figure 8, left panel). Similar results were also obtained for the in vitro studies using hepatocytes as the working model (Figure 8, right panel).

Effect on Bcl-2 family proteins

The Bcl-2 proteins are a family of proteins involved in response to apoptosis and these are upstream regulator of mitochondrial membrane potential, release of cytochrome C and subsequent activation of caspses. Because the process of apoptosis is considered to be regulated by a complex interplay of proapoptotic (Bax) and antiapoptotic (Bcl-2) mitochondrial membrane proteins, the status of these signaling molecules was also investigated in Pb(NO₃)₂ induced liver tissues of the experimental sets of animals.
Figure 8. Immunoblot analysis of NF-κB, IκBα and IKKα proteins in response to Pb(II) and mangiferin treatment, both in liver and hepatocytes. Hepatocytes were treated with Pb(II) (30 μg/mL) and mangiferin (50 μg/mL) for 2 h. β-actin was used as an internal control. Cont: normal control, Mang: treated with mangiferin, Pb(II): administered with Pb(II), Pb(II)+Mang: Mangiferin treated post to Pb(II) administration, Pb(II)&Mang: Simultaneous exposure of Pb(II) and mangiferin in hepatocytes. Data are mean ± SEM, for 6 sets per group and were analyzed by one-way ANOVA, with Student-Newman-Keuls post hoc tests. Differences were attributed at p<0.05, and homogeneous subgroups share common superscripted letters.

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Immunoblotting demonstrated that Pb(II) downregulated the Bcl-2 and upregulated the Bax proteins in liver tissue (Figure 9, left panel). Post-treatment with mangiferin on the other hand, could protect the liver by inhibiting the alterations of these proteins. Similar results were also obtained for the in vitro studies using hepatocytes as the working model (Figure 9, right panel).

**Figure 9. Immunoblot analysis of Bax and Bcl-2 in response to Pb(II) and mangiferin treatment, both in liver and hepatocytes.**

Hepatocytes were treated with Pb(II) (30 µg/mL) and mangiferin (50 µg/mL) for 2 h. β-actin was used as an internal control. Cont: normal control; Mang: treated with mangiferin, Pb(II): administered with Pb(II), Pb(II)+Mang: Mangiferin treated post to Pb(II) administration, Pb(II)&Mang: Simultaneous exposure of Pb(II) and mangiferin in hepatocytes. Data are mean ± SEM, for 6 sets per group and were analyzed by one-way ANOVA, with Student-Newman-Keuls post hoc tests. Differences were attributed at p<0.05, and homogeneous subgroups share common superscripted letters.

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**Effect on mitochondria dependent cell death pathway**

Mitochondria play a significant role in the regulation of cell death. Loss of mitochondrial membrane potential (ΔΨm) and release of cytochrome C in the cytosol are the novel biomarkers of oxidative stress induced cell damage from mitochondria and subsequent activation of initiator caspase 9 in addition to effector...
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A.

B.

Liver tissue
- Cytochrome c: 15 kD
- Apaf 1: 131 kD
- Caspase 9: 35 kD
- Caspase 3: 17 kD
- β actin: 42 kD

Hepatocytes
- Cytochrome c: 15 kD
- Apaf 1: 131 kD
- Caspase 9: 35 kD
- Caspase 3: 17 kD
- β actin: 42 kD

Graphs showing protein expression levels with treatment groups compared to control.
caspase 3 represent a key step in the mitochondrion dependent apoptotic cell death pathway. To establish whether mangiferin applies its anti-apoptotic activities against Pb(II)-induced apoptotic death via above pathway, we measured the mitochondrial membrane potential ($\Delta \Psi_{m}$) in the liver tissue as well as cytosolic cytochrome C, caspase 9, caspase 3 and Apaf1 levels in the liver tissue and hepatocytes. Results showed that Pb(II) administration considerably reduced the mitochondrial membrane potential (Figure 10A), increased the concentration of cytosolic cytochrome C, down regulated Apaf1 together with up-regulating caspase 3 and caspase 9 (Figure 10B). Post-treatment of mangiferin to the Pb(NO$_3$)$_2$ exposure could, however, significantly inhibit Pb(II)-induced changes of these parameters.

Histological assessment

Histological assessment of different liver sections of the normal and experimental animals have been presented in Figure 11. Apoptotic damage along the central vein and disorganized normal radiating pattern of cell plates around it have been observed in Pb(II)-induced liver sections. Mangiferin treatment after Pb(NO$_3$)$_2$ exposure showed a significant improvement in liver morphology.

Discussion

It has been well established that exposure to heavy metals including a number of environmental pollutants can cause cellular damages through the formation of highly reactive substances such as reactive oxygen species (ROS). ROS show a wide range of pathophysiology [77, 78]. The present study established that exposure to lead nitrate Pb(II) significantly increased ROS formation, enhanced oxidative stress and induced apoptosis in the liver tissue of experimental mice. This adverse effect of Pb(II), however, could be eliminated by mangiferin treatment probably because of its strong free radical scavenging activity. Besides dietary antioxidants, the body depends on several endogenous defense mechanisms to protect against ROS-induced cell damage. Among these antioxidant molecules, SOD and CAT jointly play important roles in the exclusion of ROS. With the purpose of removing excess free radicals from the system, GST and GPx use GSH in their course of reactions. Diminish in GSH content because of oxidative stress reduce the actions of GST and GPx with a concomitant decrease in the activity of GSH stimulating enzyme, GR. The sulfhydryl group of GSH directly binds to heavy metals due to a high affinity for sulfhydryl groups. Lead, arsenic and mercury effectively inactivate the glutathione molecule so it is unavailable as an antioxidant or as a substrate in liver metabolism [79]. In the present study, we found that Pb(II) exposure decreased the activities of the antioxidant enzymes, CAT, SOD, GST, GPX and GR in addition to the level of GSH in the liver tissue. Pb(II)-intoxication is also connected to the increased hepatic levels of lipid peroxidation and protein carbonylation and serum marker enzymes (ALT and ALP). But post treatment of animals with mangiferin after Pb(II) exposure could change the alterations in the activities of the antioxidant enzymes and the level of GSH. It also modulated the levels of lipid peroxidation and protein carbonylation and serum marker enzymes.

We studied Pb(II) induced mode of cell death and its protection by mangiferin using DNA fragmentation (in liver tissues) and flowcytometric analyses (in hepatocytes). DNA fragmentation is one of the most often used techniques in the study of cell death. Internucleosomal DNA fragmentation can be visualized by gel electrophoresis as the characteristic DNA ladder formation and was considered as a biochemical hallmark of apoptosis. In our study, DNA gel electrophoresis showed that Pb(II) exposure caused DNA fragmentation which appeared as a ladder in the agarose-ethidium bromide gel. The result of this study clearly suggests that Pb(II) induced cell death occurred via apoptotic pathway. Mangiferin could, however, inhibit the Pb(II) induced DNA fragmentation and apoptotic cell death. Flowcytometric analyses also demonstrated that Pb(II) mostly damaged hepatocytes via apoptotic pathway. Simultaneous treatment with mangiferin, on the other hand, decreased the degree of Pb(II)-induced apoptotic cell death.

Multicellular organisms have three well-characterized subfamilies (p38, ERK1/2 and JNK) of mitogen activated protein kinases (MAPKs). The members of the family are basically serine/threonine kinases, activated by dual phosphorylation on their threonine and tyrosine residues and are projected as critical redox signaling proteins. They control a vast array of physiological/pathophysiological processes involved in organ dysfunctions. To investigate the molecular mechanism underlying the protective action of mangiferine, we explored whether one or more members
of this family plays any role in Pb(II)-induced oxidative stress and cellular dysfunction in liver as well as in hepatocytes. We observed a noticeable increase in protein content of phospho-JNK, p38 and ERK (1/2) without any alteration in total protein content of these MAPKs family proteins in Pb(II)-induced liver toxicity (Figure 7, left panel). Similar results were also obtained when an in vitro study was conducted using hepatocytes (Figure 7, right panel) as the working model.

Earlier studies also suggest that in addition to MAPKs activation, NF-κB pathway is also involved in Pb(II) induced organ pathophysiology [80]. NF-κB is known to be a rapidly induced transcription factor among many involved in the stress-responsive intracellular signaling pathways and is highly sensitive to the alterations of cellular oxidative status, cell transformation, and apoptosis [81,82]. Activation of this transcription factor could be regulated by the phosphorylation of its p65 subunits and degradation of its inhibitor-κB (IκB) via phosphorylation of IκKa/b resulting its translocation into the nucleus [83]. In our study, we also found the up-regulation of the phospho NF-κB in response to Pb(II) induced liver damage and hepatocytes cytotoxicity signifying its pro-apoptotic role. These results also supported the fact in the existing literature. Mangiferin, on the other hand, successfully suppressed the Pb(II) induced up-regulation of MAPKs family proteins and phospho NF-κB both in vivo and in vitro. So, it can be concluded that at least a part of the beneficial effects of mangiferin in Pb(II) induced hepatic pathophysiology is due to the inhibition of the MAPKs-NF-κB pathways.

There exist a balance between the proapoptotic (Bax/Bad) and antiapoptotic (Bcl-2, Bcl-xl) members of the Bcl-2 family proteins and their up and down regulations usually determine the fate of the cells either to undergo apoptosis or to survive in an organ pathophysiology. In addition, these proteins are the upstream regulators of mitochondrial membrane potential (ΔΨm) and release of cytochrome C into cytosol. Mitochondria play an important role in apoptosis or programmed cell death pathway. A number of studies suggest that the change in mitochondrial membrane potential is able to switch the committed cells to apoptotic death with oxidative stress as the mediator [84]. Throughout this process, the electrochemical gradient across the mitochondrial membrane changes, leading to a decrease in ΔΨm. This decrease in ΔΨm results in the release of cytochrome C from the mitochondria into the cytosol, where it activates the caspase cascade, leading to programmed cell death.

![Figure 12. Reaction of mangiferin with free radicals (ROS) and resonating canonical structures of two phenoxyl radicals.](doi:10.1371/journal.pone.0056894.g012)

![Figure 13. Schematic representation of Pb(II) induced hepatotoxicity and its protection by mangiferin.](doi:10.1371/journal.pone.0056894.g013)
membrane collapse. Mitochondria have been expressed as the sensor of oxidative stress and loss of its membrane potential ($\Delta V_m$) or formation of a pore in the mitochondrial membrane (called the Permeability Transition pore, or PT pore) all together can show the way of cell death through the release of cytochrome C. Once cytochrome C is released into the cytosol it is able to interact with a protein called Apaf-1. This leads to the recruitment of pro-caspase 9 into a multi-protein complex with cytochrome C and Apaf-1 called the apoptosome. Formation of the apoptosome leads to activation of initiator caspase (caspase 9) as well as the effector caspase (caspase 3) and induces apoptosis. In the present study, we found that Pb(II) up regulated the expression of Bax in addition to a down regulation of the expression of Bcl-2 in both the liver tissue and hepatocytes, reduced the mitochondrial membrane potential, enhanced the release of cytochrome C in the cytosol, down regulated Apaf-1 and activated caspases (caspase 3 and caspase 9) both in vivo and in vitro. Post treatment with mangiferin, on the other hand, successfully suppressed all these mitochondrial dependent apoptotic events in Pb(II) induced hepatic damage suggesting its protective action in this particular pathophysiology. 

From the structural point of view, it is evident that mangiferin contains four polyphenolic H-atoms. Two of them could easily be abstracted by suitable free radicals to form two stable phenoxy radicals (Figure 12) and this property of mangiferin probably explain its free radical scavenging activity [65].

Conclusions

In conclusion, results from our study revealed that Pb(II) not only activates NF-kB activation via IKK pathway in the liver tissue but also remains responsible for the increased phosphorylation of MAPKs and ultimately leads to hepatic cellular apoptosis via mitochondria dependent pathway. Mangiferin, on the other hand, could act as a protective agent in this Pb(II) induced pathophysiology by enhancing antioxidant defense and acting through the mitochondrial dependent as well as via the inhibition of MAPKs and NF-kB pathways (Figure 13). In other words, mangiferin supplementation appears to be a promising approach for the hepatoprotection in Pb(II)-induced liver dysfunction and cell death. This xanthone, therefore, deserves further research as a potent beneficial agent in hepatic and other organ pathophysiology because of the absence of any noticeable toxicity and its multiple advantageous properties.

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Author Contributions

Edited manuscript: PBP KS PCS. Conceived and designed the experiments: PBP KS PCS. Performed the experiments: PBP KS PCS. Analyzed the data: PBP KS PCS. Contributed reagents/materials/analysis tools: PBP KS PCS. Wrote the paper: PBP KS PCS.

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