Nanoscale structures formed in silicon cleavage studied with large-scale electronic structure calculations; surface reconstruction, step and bending.

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The 10-nm-scale structure in silicon cleavage is studied by the quantum mechanical calculations for large-scale electronic structure. The cleavage process on the order of 10 ps shows surface reconstruction and step formation. These processes are studied by analyzing electronic freedom and compared with STM experiments. The discussion presents the stability mechanism of the experimentally observed mode, the (111)-(2 × 1) mode, beyond the traditional approach with surface energy. Moreover, in several results, the cleavage path is bent into the experimentally observed planes, owing to the relative stability among different cleavage modes. Finally, several common aspects between cleavage and other phenomena are discussed from the viewpoints of the nonequilibrium process and the 10-nm-scale structure.

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I. INTRODUCTION

Cleavage is a nonequilibrium process and its dynamical mechanism is essential. In particular, the cleavage of silicon single crystals is of great interest from the multiscale viewpoint between macroscale and atomic scale pictures. In the macroscale picture, silicon shows perfect brittleness and brittle fracture is usually described by continuum mechanics.$^{1,2,3}$ In the atomic scale picture, on the other hand, a cleaved surface contains areas with well-defined reconstructions. Currently, these atomic scale structures are observed by scanning tunneling microscopy (STM)$^4$ and other experiments. The multiscale feature of the phenomenon, as discussed in this paper, appears in nanoscale (or 10 nm scale) processes near the crack tip, though such processes cannot be seen by direct (in situ) experimental observation.

A fundamental question is what Miller index and surface reconstruction appear at the cleavage surface. A traditional prediction is that the cleavage plane should be that with the smallest surface energy, or the smallest energy loss with forming surface. This prediction, however, is actually not satisfactory owing to the following experimental facts; (i) The easiest cleaved surface of Si is a metastable (111)-(2 × 1) structure,$^{4,5,6,7,8,9,10}$ and will change, irreversibly, to the ground-state (7 × 7) structure.$^4$ (ii) The (110) cleavage plane is also experimentally observed but less favorable (see an experimental (STM) study,$^2$ and a recent theoretical study$^{13}$). (iii) The cleaved Ge(111) surface also shows the same (2 × 1) structure, while the ground state surface structure is the c(2×8) structure.$^{14,15}$ These facts imply the importance of direct cleavage simulations with electronic structure. Although such simulations have been carried out thus far,$^8,9,13$ the investigation is still limited, owing to the system size of 10$^5$ atoms.

In this paper, the cleavage of silicon is studied with quantum mechanical calculations for large-scale electronic structure,$^{16,17,18,19,20,21}$ and we use a transferable Hamiltonian$^{22}$ in the Slater-Koster (tight-binding) form. The methodology is reviewed briefly in Appendix A. The method realizes the cleavage process with more than 10$^4$ atoms or a sample length of 10 nm. This paper is organized as follows; Section II describes the important aspects discussed in this paper. The easiest cleavage mode on the (111)-(2 × 1) plane is discussed in Section III and Section IV. The latter section focuses on step formations. The simulation results for bending in the cleavage path are presented in Section V. Finally, in Section VI several common aspects of nanoscale structures are discussed for cleavage and other phenomena.

FIG. 1: (a)-(c): Toy model of cleavage, in which surface reconstruction is illustrated as dimerization. (d): Possible cleavage paths on Si(111) plane.
II. ASPECTS OF CLEAVAGE PROCESS IN REAL CRYSTAL

The first aspect of cleavage is the typical time scale as a nonequilibrium process. The time scale is determined by the cleavage propagation velocity. In the continuum mechanics and many experiments, the propagation velocity is given on the order of, but less than, the sound velocity or the Rayleigh wave velocity ($v_R = 4.5 \text{ km/s} = 4.5 \text{ nm/ps}$ for Si). Since the atomistic process occurs within the time scale, the reconstruction in cleavage, unlike that in annealing, should occur locally. In a typical process, a surface-bound state is formed between electrons in nearest-neighbor dangling bond sites, as illustrated in Figs. 1(a)-(c). In other words, the elementary process should contain two successive bond breakings, not one. This nearest-neighbor reconstruction mechanism will directly give the experimentally observed (2×1) reconstruction (See Section III).

The second aspect comes from the crystal structure. Figure 1(d) shows possible cleavage paths on the Si(111) plane, with or without step formation. Unlike the toy model of Figs. 1(a)-(c), the system does not have the mirror symmetry with the cleavage plane and the upper and lower cleaved surfaces are inequivalent in symmetry. In particular, the inequivalence between the upper and lower stepped paths in Fig. 1(d) is distinctive in experiments.

Since experiments reported only the (111) or (110) cleavage plane, theory should explain why other surfaces do not appear. An interesting fact is that the calculated surface energy of the reconstructed (001) surface is smaller than that of the (111)-(2×1) surface ($\gamma_{(001)} = 1.41 \text{ J/m}^2 < \gamma_{(111)}^{2\times1} = 1.44 \text{ J/m}^2$). This means that the absence of the (001) cleavage surface is not simply predicted by surface energy. The possibility of the (100) cleavage mode was investigated in our previous study in which the (001) cleavage planes are observed for small sample sizes less than 10 nm. For larger sizes, the flat (001) cleavage surface becomes fairly unstable and many steps form. The instability of the (001) cleavage mode is helpful in understanding the stability of the (111) cleavage mode, as discussed in Section IV.

III. (111)-(2×1) CLEAVAGE MODE

In experiments, the cleavage on the Si(111)-(2×1) plane is the easiest cleavage mode. As the atomic structure of the cleaved surface, Pandey’s π-bonded structure is now widely accepted. An actual surface formation process was shown when a parallel separation is introduced in a slab with the minimal periodic simulation cell for the (2×1) structure (See Figs. 1 and 2 of Ref. 26).

The present cleavage simulations are performed with the [211] propagation directions (Fig. 1(d) for geometry), which are consistent with those in typical experiments (See Fig. 1 of Ref. 26, for example). An external load is imposed and its physical origin is the concentrated elastic field in the macroscale experimental sample. See Appendix B for details. A smaller sample with 416 atoms was also simulated as shown in Appendix C so as to confirm the quantitative agreement with previous experimental or theoretical studies. In all cases, the periodic boundary condition is imposed in the [01\bar{1}] direction, which is orthogonal to the cleavage propagation direction. The periodic length contains eight atomic layers or is four times larger than that of the minimum.
crystalline periodicity. In some simulations, the atomic motion is under a constraint by the minimum crystalline periodicity in the [01\bar{1}] direction. We call this constraint, the ‘2D-like constraint’. For a systematic investigation, simulations were carried out with and without the ‘2D-like constraint’.

Figure 2 shows an example of the simulation within the ‘2D-like constraint’, in which a step appears. The sam-
ple contains 11,096 atoms. Hereafter, the bonds (rods) are drawn just as a guide for eye. From Fig. 2, the cleavage propagation velocity is estimated to be \( v_{\text{prop}} \approx 2\text{nm/ps}=2\text{km/s} \), as expected in Section II. The cleaved surface in (b) and the latter appears in (d). The oval in (e) indicates the presence of a lone pair state on the D atom site. The oval in (f) indicates the π-bonded zigzag chain in the direction perpendicular to the page. (g)-(i): Quantum mechanical analysis of a process in which the bonding wavefunction (\( \phi_i \)) between A and B sites changes into that between A and C sites; (g) The one-electron energy \( \varepsilon_i \equiv \langle \phi_i | H | \phi_i \rangle \) and the weight of s orbitals \( f_{s}^{(i)} \) for the wavefunction \( \phi_i \). (h) The atomic distance between A and B sites (\( d_{AB} \)) and that between A and C sites (\( d_{AC} \)). (i) The spatial weight distribution on the A, B and C atom sites (\( n_A, n_B, n_C \)) for the wavefunction \( \phi_i \). The rest of the weight \( (1-n_A-n_B-n_C) \) is also plotted.

A. Elementary reconstruction process

Figures 3(a)-(d) show an example of (2 \times 1) reconstruction. As in the previous simulation with the parallel separation, the reconstruction process has two stages; the buckled (2 \times 1) structure of Fig. 3(e) appears, as shown in Fig. 3(b), when the two dangling-bond elec-
trons at the C and D atom sites form an atomic (lone-
pair) state at the D atom site\(^\text{27}\). After that, the π-bonded (2 \times 1) structure of Fig. 3(f) appears, since (i) the atomic state is transformed into the π-bonding state between the B and D sites and (ii) the bonding state between the A and B sites is transformed into one between the A and C sites, so as to reproduce the tetrahedral coordination in the C site. In the three-dimensional view, a π-bonding zigzag chain appears in the direction perpendicular to the page. Hereafter, the transition from the buckled structure into the π-bonded one is called the ‘BP transition’. Note that the resultant π-bonded structures are always tilted, since the D or B atom site is shifted to the vacuum side. The two tilted structures are inequivalent in symmetry and have a small difference in surface energy.\(^{10,14}\) In this paper, however, we do not focus on the difference between the two tilted structures.

The above explanation of (ii) is confirmed by the fol-
lowing quantum mechanical analysis; Figure 3(g) shows the one-electron energy \( \varepsilon_i \equiv \langle \phi_i | H | \phi_i \rangle \) and the weight of s orbitals \( f_{s}^{(i)} \)\(^\text{18}\). For example, \( f_{s}^{(i)} = 1/4 \) in an ideal \( sp^3 \) hybridized state. Figure 3(h) shows the distances between sites A, B and C. Figure 3(i) shows the spa-
tial weight distribution (\( |\phi_i(r)|^2 \)) on atom sites. Note that the intermediate wavefunction, indicated as the arrow in Fig. 3(h) or (i), shows no characteristic feature in \( \varepsilon_i \) and \( f_{s}^{(i)} \), such as maxima, minima or plateaus. This is quite different from the reconstruction process on the (001) surface\(^\text{26}\) in which the intermediate wavefunction shows a plateau with a large weight on the s orbitals (\( f_{s}^{(i)} \approx 0.8 \)).
and lower surfaces.

We also found that the sample geometry can have an effect on the possible BP transition, through strain. For example, the BP transition seems to occur more easily in a thicker sample, with a larger sample length in the [111] direction, since the BP transition should accompany the strain in deeper (subsurface) layers. This observation was unchanged, when additional simulations were carried out by more complicated Hamiltonians with a better reproduction of the surface energy. A more systematic investigation should be carried out, in the future, on the quantitative condition of the BP transition.

IV. STEP FORMATION IN CLEAVAGE

Several (111) cleavage simulations contain step formation. There are two inequivalent paths shown as the upper and lower paths in Fig. 1(d). The ‘upper’ type of path is usually called the [211] type, and the ‘lower’ type is called the [2 ¯11] type. The two types of step are experimentally reported. In early papers, it was reported that the lower path in Fig. 1(d), the [211] type is predominant. In later experimental papers, however, the upper path, i.e. the [211] type step, was reported. In our simulations, both types of step are observed.

A. ‘Upper’-type step

First, we discuss the step in Fig. 2 which is classified into the ‘upper’ path in Fig. 1(d) or the [211] type. This type of step was observed with and without the ‘2D-like’ constraint. The formation process is shown in Fig. 4. Hereafter, the members of several rings are plotted, such as ‘6’, ‘5’ or ‘7’, so as to clarify the reconstruction. This step can be decomposed into the successive bendings in the cleavage path as (111) → (1 ¯11) → (111) planes (See Fig. 1(d) and Fig. 2). That is, the step is formed through the (111) plane and we call the step, the ‘via-(111)-plane’ type. In the crystalline geometry, the (111) and (1 ¯11) planes are equivalent. If we ignore the reconstruction (or quantum mechanical) freedom, the step formation can be understood, since bending between equivalent planes can occur by the local fluctuation of the (concentrated) elastic field at the crack tip, particularly in its angular dependence.

As a quantum-mechanical analysis, snapshots of the step formation are shown in Fig. 5 in which an atom with an excess or deficit electron population is marked ‘+’ or ‘−’, respectively. The initial bonding states (dashed line) changed into the surface-bound state mainly localized on a ‘+’ atom site. An atom ‘+’ is always placed at the vacuum side of a buckled structure, which can be understood by the general quantum mechanical tendency. The resultant surface shows a balanced structure, owing to the alternate alignment of the ‘+’ and ‘−’ sites. The
The present step structure is concluded to be a possible one, although it has not yet been confirmed experimentally.

Now we comment on a pioneering theoretical paper, by Chadi and Chelikowsky, in which the ‘upper’ and ‘lower’ types of step are compared with assumed atomic structures. They concluded that the upper path is unrealistic, which seems to be in contrast to the present result. The contrast appears, because the present step structure is different from that assumed in Ref. They concluded that the upper path is unrealistic, which seems to be in contrast to the present result. In this reference, the dangling bonds at the step edge are assumed to be rebonded with each other in the perpendicular direction of the page of Fig. (d), but such a rebonding process results in only a tiny energy gain. In the present step structure of Fig. on the other hand, the dangling bond electrons at the step edge are rebonded within the plane of Fig. (d), whose energy gain mechanism is explained above.

B. ‘Lower’-type step

Several simulations result in the appearance of the other type of step, the lower-type step in Fig. (d) or the [211] type, when they were carried out without the 2D-like constraint or minimal periodicity in the [01\bar{1}] direction. An example is shown in Fig. (a). To observe the freedom in the [01\bar{1}] direction, we classify the atoms, by their initial positions, into subsystems or ‘slices’. Each ‘slice’ contains atoms within the minimum unit length (two atom layers) in the [01\bar{1}] direction. Four slices are defined in Fig. (a) and two of them are shown in Figs. (b) and (c).

In Fig. (a), the cleaved surface is defective before the step formation. In the defective area of Figs. (b) and (c), six-membered rings appear in the surface layer, as well as five- or seven-membered rings. A defective area also appears in flat (nonstepped) areas, as in Fig. (a). The appearance of the defective six-membered rings means that the reconstruction of the dangling-bond electrons does not occur within the slice. We should recall that an ideal (111) surface has symmetry with a $\pm 2\pi/3$ rotation and the (2 x 1) reconstruction mechanism shown in Fig. (f) or (g) is possible in the [2\bar{1}1], [\bar{1}2\bar{1}] and [1\bar{1}2] directions. This type of step formation is decomposed into the successive bendings of cleavage as (111) $\rightarrow$ (100) $\rightarrow$ (111) planes. In other words, the step is formed through the (100) plane and we call the step the ‘via-(100)-plane’ type (See Fig. (d) for geometry). At the step edge, the dangling bond sites are equivalent to those on a (100) or (001) plane. The reconstruction among them is possible with an energy gain of dimerization. The actual process of successive bond breakings and reconstructions on (100) surface can be obtained as an unstable cleavage mode. With the 2D-like constraint, however, the dimerization is prohibited. This is why this type of step does not appear within the 2D-like constraint.

The geometries in Fig. (a) can be compared with those proposed in STM experiments. The geometry of Fig. (b) is identical to that of Fig. (b) in Ref. because of the same alignment of the ring structures; $5 \rightarrow 7 \rightarrow 6 \rightarrow 5$ before the step and $6 \rightarrow 5 \rightarrow 7$ after the step. In the same manner, the geometry of Fig. (c) is identical to that of Fig. (a) in Ref. only before the step ($5 \rightarrow 7 \rightarrow 5$). After the step, however, the present geometry ($6 \rightarrow 5 \rightarrow 7$) is different, by one six-membered ring, from that shown in Ref. ($6 \rightarrow 6 \rightarrow 5 \rightarrow 7$).

The present investigation does not show that the upper- or lower-type of step in Fig. (d) is more favorable than the other, because the present simulations were carried out under the periodic boundary conditions in the [01\bar{1}] direction. We speculate that the appearance of the upper path, the [211]-type step, should be enhanced, because the step can be formed within the minimal crys-
talline periodicity in the [01\textbar] direction. We have been informed that the two types of step are observed in STM images [26] and we consider that a more systematic investigation is desirable both theoretically and experimentally.

C. Step formation and stability of cleavage mode

The present study of step formation can lead to the stability mechanisms of the (111)-(2 × 1) cleavage mode, since a stable cleavage mode should be robust against possible disorders and fluctuations. We have discussed that a step formation is decomposed into successive bendings of cleavage plane, such as (111) → (100) → (111) planes. The former and latter bending processes correspond to the deviation and (quick) recovery of the (111) cleavage mode, respectively. Since the π-bonded (2 × 1) structure has a π-bonded zig-zag chain perpendicular to the cleavage propagation direction, it can accompany the ordering in the surface structure. Actually, in Fig. 4 the step formation changes the cleaved surface from a disordered (defective) structure into an ordered (2 × 1) structure.

Another stability mechanism can be found in the possibility of multiple propagation directions. When the (2 × 1) structures on the (111) and (001) surfaces are compared, a crucial difference comes from their symmetry. In the limited growth of the (001) cleavage mode, the cleavage propagates easily in the dimerization direction. The axis of the dimerization direction is unique on the (001) surface and a (single) step formation changes the axis perpendicularly ((2 × 1) → (1 × 2)). As a consequence, a cleavage growth with multiple propagation directions requires step formation and the resultant (001) cleaved surfaces tend to be rough with many steps [26]. The steps on the (111) surface, on the other hand, can preserve the cleavage propagation direction or the direction of the π-bonded chain ((2 × 1) → (2 × 1)), as shown in this section. Moreover, the ideal (111) surface, unlike the (001) surface, has three equivalent axes of stable cleavage propagation; the [21\textbar], [12\textbar] and [11\textbar] axes. In conclusion, the stability of the (111) cleavage mode is supported by the geometrical features; (i) A step can be formed without changing the cleavage propagation direction. (ii) The cleavage propagation direction can be changed without step formation. The above feature (ii) should result in multiple domains without step formation. Actually, an experimental STM image, Fig. 2 on Ref. [26], shows a cleaved (111) surface with multiple domains. We speculate that such a multiple domain structure is formed, because the cleavage propagation directions are locally different.

We should emphasize that the present stability mechanisms, the robustness against step formation and the possible growth with multiple propagation directions, cannot be explained by traditional discussion on surface energy.

V. BENDING IN CLEAVAGE PATH

Finally, a direct investigation was carried out for the preference of the (111) or (110) surface in the cleavage process. We expect that, even if the simulation initiates the cleavage on other planes, the cleavage path will be bent into a more favorable plane. The above was confirmed by simulations. Hereafter, symmetrically equivalent planes are called by their ‘type’. For example, (1\textbar\textbar) and (11\textbar) planes are (11\textbar)-type planes.

An example is shown in Figs. 5(a)-(e). The sample contains 10368 atoms and the sample size is 10nm × 14nm on the (110) plane. The periodic boundary is imposed in the [10\textbar] direction by eight atomic layers. We prepare an initial ‘seed’ of cleavage on the (001) plane. An external force acts on the atoms near the sample surfaces except the left one. See Appendix B for details.

Figure 5(f) shows the broken-bond sites on the crystalline geometry and the corresponding indices are drawn in Fig. 5(g). In this case, the bendings are observed from the initial (001) plane to (11\textbar)-type and/or (11\textbar)-type planes. The actually observed path shows the bending of (001) → (11\textbar) → (11\textbar) planes. Here, the (11\textbar)-type planes appear only near the sample surface, which implies that the appearance of (11\textbar)-type planes is enhanced by the present sample geometry. Different paths were observed under different simulation conditions, such as tuning the magnitude of the external load, preparing different sample geometry, changing the region of imposing external force and setting a different cleavage seed. For example, we observed a bending between two (11\textbar)-type planes, such as (11\textbar) → (11\textbar) planes. In conclu-
 FIG. 8: Cleavage process of silicon with bending of cleavage planes. (a)-(e): Snapshots with a time interval of approximately 1 ps. Atoms are drawn as balls from the viewpoint of the [110] direction. (f): The broken bond sites in snapshot (e) are plotted as bold lines in the ideal (crystalline) geometry. (g) Indices of geometry in (f).

sion, a (111)-type or (110)-type plane appears, while no (001)-type plane appears except the initially prepared one. This conclusion is consistent with the experimental preference of the (111)- and (110)-type cleaved surfaces.

Moreover, the bending path changes, sometimes, from a disordered (defective) area into an ordered one with a well-defined reconstruction; In Fig. 9(a), the (110) plane after bending contains the buckled structures, known for typical reconstruction. In the process shown in Figs. 9(b)-(c), the resultant (111) cleavage plane contains the $\pi$-bonded (111)-(2 x 1) structure.

VI. GENERAL DISCUSSION

The present investigation of cleavage can be discussed from the general viewpoints of (I) time scale and (II) length scale. From the viewpoint of time scale (I), the metastable Si(111)-(2 x 1) structure appears, owing to the local reconstruction mechanism within a limited time scale. Since the mechanism is general in the quantum mechanical picture, we expect the (111)-(2 x 1) surface also in other nonequilibrium processes. Actually, we were informed that the corresponding STM image is found in a Si(111)$\sqrt{3} \times \sqrt{3}$-Ga surface, when the Ga atoms are locally removed at room temperature. At higher temperatures, the (7 x 7) structure appears. The above situation is another ‘fast’ process almost free from thermal equilibration. A review article lists other experiments with the appearance of the Si(111)-(2 x 1) structure.

From the viewpoint of length scale (II), structure of material is generally determined by volume and surface terms of energy. Dimensional analysis shows a crossover of mechanical property between nano- and macroscale systems: The two energy terms are competitive in nanoscale system, while the volume term is dominant in macroscale system. In the case of silicon nanostructure, the above energy competition originates from that between the $sp^3$ (bulk) term and the non-$sp^3$ (surface) term. In the present investigations, the crossover at the 10 nm scale is found in step formation or bending in cleavage path. We speculate that the crossover at the 10 nm scale is universal in silicon, because the surface energy is always on the same order among different surface indices and reconstructions. The crossover may be seen also in the size dependence of the shape in self-organized Si islands. Islands with a size of 10 nm or less have a semispherical shape and those with a larger size have a pyramidal shape with facets in well-defined indices. These crossovers at the 10 nm scale can be understood, because a well-defined reconstructed surface appears, only when the system size is much larger than the unit length of the reconstructed surface (~1 nm).

The present method of large-scale electronic structure calculation has wide applications and is not specific to cleavage. The structural property at the 10 nm scale is an urgent problem of the present technology and can be a future study of the present method.

APPENDIX A: METHODS FOR LARGE-SCALE ELECTRONIC STRUCTURE CALCULATION

In recent years, we have developed theories and program codes for large-scale electronic structure calculations. Among them, a variational (VR) procedure for generalized Wannier states is used in the present cleavage simulations. The Wannier state is well-defined localized ‘chemical’ wavefunction in condensed matter, such as a bonding state or a lone-pair state, with a slight spatial extension. The suffix $i$ of a Wannier state $\phi_i$ denotes its localization center. A benchmark is shown in the left panel of Fig. 10, in which our calculations are compared with the conventional method for calculating eigenstates. The figure shows not only the result of the VR procedure but also that of another Wannier state method called ‘perturba-
for decades\textsuperscript{27,38,39} and can be founded by the \textit{ab initio} theory\textsuperscript{30,41,42}. Consequently, group IV elements can be systematically described by a one-parameter energy scaling theory\textsuperscript{16,17,37,38}. The scaling parameter, ‘metallicity’ $\alpha_m$, is defined as

$$\alpha_m = \frac{\varepsilon_p - \varepsilon_s}{2t_{sp^3}}.$$  

Here, $(\varepsilon_p - \varepsilon_s)$ is the energy difference between the $p$ and $s$ orbitals and $t_{sp^3}$ is the transfer energy along a bulk $(sp^3)$ bond ($t_{sp^3} = |V_{ssp} - 2\sqrt{3}V_{sp\sigma} - 3V_{pp\sigma}|/4$). A typical value for C is $\alpha_m = 0.35$.\textsuperscript{33} The values for Si and Ge are similar, $\alpha_m = 0.75 - 0.78$ (see Refs.\textsuperscript{16,17} and references therein).

Several surface structures among C, Si and Ge are systematically explained by the energy scaling theory. An example is shown in the dimer geometry on (001) surfaces. \textit{Ab initio} calculations result in a \textit{symmetric} dimer in the C case and similar \textit{asymmetric} dimers in the Si and Ge cases (see Fig. 2 of Ref.\textsuperscript{15}, for example). In the right panel of Fig. 10, the above trend is reproduced with the present Hamiltonian by tuning the value of $\alpha_m$ or $(\varepsilon_p - \varepsilon_s)$. Since the (111)-(2 $\times$ 1) cleaved surface is commonly seen in Si and Ge\textsuperscript{15}, the phenomena should hold a common mechanism in the present context of universality.

For validation of the present study, the surface energy of several Si surfaces was calculated by the present Hamiltonian as follows and the \textit{ab initio} values\textsuperscript{14} are given inside the parentheses: $\gamma_{001}^{(001)} = 1.58 (1.41) \text{ J/m}^2$, $\gamma_{111}^{(111)} = 1.97 (1.44) \text{ J/m}^2$ and $\gamma_{110}^{(110)} = 2.11 (1.70) \text{ J/m}^2$, for the (001)-(c(4 $\times$ 2)) structure, the $\pi$-bonded (111)-(2 $\times$ 1) structure and the buckled (110) structure, respectively. The present Hamiltonian reproduces the \textit{ab initio} results satisfactorily, particularly, the order of magnitude ($\gamma_{111}^{(111)} < \gamma_{110}^{(110)} < \gamma_{110}^{(buckled)}$) though the absolute values are somewhat overestimated. The calculated surface energy of (001) surfaces also gives satisfactory results\textsuperscript{35}.

\section*{APPENDIX B: CONDITIONS OF CLEAVAGE SIMULATION}

This appendix explains the technical conditions for cleavage simulation. The time step of the molecular dynamics was set at 3 fs. The center of gravity of the sample was fixed as a constraint. Sample surfaces were terminated by orientationally fixed $sp^3$ bonding states. This boundary condition corresponds to the situation in which the sample is embedded in a bulk ($sp^3$-bonded) system. This situation is quite different from that with usual hydrogen termination, in which the atomic structure is deformed significantly from the tetrahedral geometry, owing to the large deviation from the $sp^3$ bonding.

Initial defect bonds were prepared for the cleavage ‘seed’, which can be seen in Fig. 2(b) or Fig. 8(b). Actually, an additional short-range repulsive force was imposed on the atom pairs of selected bond sites. The se-
The selected layers is 5 or 10 % of the sample length. The sample surfaces, except the left one. The thickness of the force field is imposed on selected atomic layers near the sample boundary in Fig. 2 (a)-(e).

The details are described elsewhere. Within the above requirements for the artificial movement, the conclusions of the present paper are not changed. The way of concerning the artificial movement is introduced to the sample surfaces with the characteristic velocity. In our simulations, the artificial movement is based on the results that are not sensitive to the boundary conditions. We pick out the cases of Figs. 2 and 8. In Fig. 2, the external load is realized by the constrained movement. The atoms of the outermost two layers on the top and bottom sample surfaces were constrained under artificial movement in the [111] direction. Since the cleavage propagation velocity is on the order of the Rayleigh wave velocity \( v_R = 4.5 \text{nm/ps} \), the velocity of the artificial movement should be much slower than \( v_R \). If not, the atoms in the constraint motion (the atoms near the sample surfaces) are removed from the sample and no internal (cleavage) surface appears. In Fig. 2, for example, the constraint movement was required to avoid forming multiple cleavage planes. The details are described elsewhere. Within the above requirements for the artificial movement, the conclusions of the present paper are not changed. The way of controlling the artificial movement is reflected on the shape of the sample boundary in Fig. 2 (a)-(e).

Now we explain the case of Fig. 8 in which an external force field is imposed on selected atomic layers near the sample surfaces, except the left one. The thickness of the selected layers is 5 or 10 % of the sample length. The force field is given as

\[
F(r) = \frac{K d_0^2}{\sqrt{2\pi r}} \frac{r}{|r|} 
\]

with the relative coordinate \( r \) from the force center. The (two-dimensional) coordinate \( r \) is given on the (011) plane (See Fig. 1). The force center \( (r = 0) \) is fixed at the position of the initial defect and is placed almost at the middle of the left sample surface. The length \( d_0 \) is determined so that the volume per atom is given as \( d_0^3 \) \((d_0 \approx 3\text{Å})\). The force field is consistent with the concentrated stress field given by the continuum mechanics of cleavage when its angular dependence is ignored. The angular dependence was ignored in our simulations, because it is defined only for a planar crack and is not valid in the stepped or bent cleavage paths discussed in this paper. In Fig. 8, \( K = 0.8\text{MPa}\sqrt{\text{m}} \), which is slightly larger than the critical value \( (K_{\text{prop}} = 0.7\text{MPa}\sqrt{\text{m}}) \) estimated in Appendix C. In several other simulations, the field center was under artificial movement, owing to the fact that the crack tip moves with the cleavage propagation. We did not find, however, a systematic difference in the resultant cleavage behavior.

Finally, in Figs. 2 and 8 the total kinetic energy was controlled by the thermostat method\( ^{22} \) with a typical temperature parameter of \( T = 800\text{K} \). The temperature parameter \( T \), however, does not correspond to experimental temperature, since the simulation is a nonequilibrium process. In Fig. 2 for example, the constraint movement is introduced to the sample surfaces with the characteristic velocity \( v_0 \). The temperature parameter \( T \) should be sufficiently large to propagate the introduced deformation into the internal region.

**APPENDIX C: ESTIMATION OF CRITICAL STRESS INTENSITY FACTOR**

The critical stress intensity factor for cleavage propagation, \( K_{\text{prop}} \), was estimated for the (111)-(2×1) cleavage mode with a smaller sample of 416 atoms. In the continuum theory with linear elasticity\( ^{24} \), the divergent stress field is presented at the crack tip. The center of the divergent field \( R_c \) is given and the stress intensity factor \( K \) is introduced as the amplitude of the divergent field. Its critical value \( K_{\text{prop}} \) is a measurement of the critical external load and was determined in several experimental techniques and in the electronic structure calculations with hundreds of atoms. \(^{25,13}\)

The present estimation of the critical factor \( K_{\text{prop}} \) was carried out within the following quasi-static picture; when the field center \( R_c \) is shifted by one atomic increment, the crack tip should be shifted, also by one atomic increment, with the elementary bond-breaking and reconstruction process. If the factor \( K \) is not sufficiently large \((K < K_{\text{prop}})\), however, the crack tip is not shifted. In practical simulations, the field center \( R_c \) was chosen...
in the middle of a bond layer, which is shown as the non-stepped path of Fig. 1(d). As constraints, outer atoms were fixed with the linear elastic displacement of plane strain, determined by the factor $K$ and the position $R_e$. Internal (movable) atoms are relaxed in the finite temperature of 300 K for 2.4 ps. After the relaxation of the internal atoms, the position $R_e$ is shifted by 1/10 of one atomic increment and the relaxation is restarted. The resultant cleaved surface shows the buckled (2 × 1) structure, as in Fig. 2 of Ref. 14. In our simulations, the increase of the reconstructed surface area by one atomic increment occurs within the typical time scale of 0.1 ps.

The present calculations resulted in a typical value of $K_{\text{prop}} = 0.7 \text{MPa} \sqrt{\text{m}}$, which agrees satisfactorily with the previous theoretical and experimental values of $0.41 - 1.24 \text{MPa} \sqrt{\text{m}}$ listed in Table I of Ref. 14. Since the calculated surface energy in Appendix A is somewhat overestimated, for example, by approximately 40% in $\gamma_{111}^{2 \times 1}$, we speculate that the present value of $K_{\text{prop}}$ is an overestimated one, owing to the limited freedom of the present Hamiltonian.

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