Synthesis and biological evaluation of some novel 1-substituted fentanyl analogs in Swiss albino mice

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ABSTRACT

Fentanyl [N-(1-phenethyl-4-piperidinyl)propionanilide] is a potent opioid analgesic agent, but has a narrow therapeutic index. We reported earlier on the synthesis and bioefficacy of fentanyl and its 1-substituted analogs (1–4) in mice. Here we report the synthesis and biological evaluation of four additional analogs, viz. N-isopropyl-3-(4-(N-phenylpropionamido)piperidin-1-yl)propanamide (5), N-t-butyl-3-(4-(N-phenylpropionamido)piperidin-1-yl)propanamide (6), isopropyl 2-[4-(N-phenylpropionamido)piperidin-1-yl]propionate (7) and t-butyl 2-[4-(N-phenylpropionamido)piperidin-1-yl]propionate (8). The median lethal dose (LD50) determined by intravenous, intraperitoneal and oral routes suggests these analogs to be comparatively less toxic than fentanyl. On the basis of observational assessment on spontaneous activities of the central, peripheral, and autonomic nervous systems, all the analogs were found to be similar to fentanyl. Naloxone hydrochloride abolished the neurotoxic effects of these analogs, thereby ascertaining their opioid receptor-mediated effects. All the analogs displayed significant analgesic effects, measured by formalin-induced hind paw licking and tail immersion tests at their respective median effective dose (ED50). They also exhibited 8–12 fold increase in therapeutic index over fentanyl. However, 5 and 6 alone produced lower ED50 (20.5 and 21.0 μg/kg, respectively) and higher potency ratio (1.37 and 1.33, respectively) compared to fentanyl. They could thus be considered for further studies on pain management.

KEY WORDS: fentanyl analogs; opioids; synthesis; toxicity; efficacy

Introduction

Severe and chronic pain conditions are usually alleviated by narcotic (opioid) analgesics like morphine (Vogel, 2002). However, their beneficial effects are often obscured by unwanted effects like constipation, nausea, vomiting, etc. (Kalso et al., 2004). Fentanyl [N-(1-phenethyl-4-piperidinyl)propionanilide] is a fully synthetic opioid analgesic that has found several clinical applications because of its rapid onset of action and good safety margin. As an analgesic, it is several times more potent than morphine (Mather, 1983; Mićović et al., 2000; Van Nimmen et al., 2004). Fentanyl is characterized by high lipophilicity and therefore it penetrates easily the central nervous system (CNS) to interact with μ-opioid receptors (mu opioid receptors; MOR), resulting in inhibition of pain neurotransmission (Kieffer, 1995; Mayes & Ferrone, 2006; Kieffer & Evans, 2009). This pharmacological characteristic of fentanyl prompted several workers to synthesize numerous new analogs of fentanyl including sufentanil, alfentanil, lofentanil, etc. (Lemmens, 1995; Scholz et al., 1996).
death (Gardocki & Yelnosky, 1964). In order to discover an analgesic with improved pharmacodynamics and pharmacokinetics, extensive efforts have been made in synthesizing several new fentanyl analogs and determining their structure activity relationship (SAR) (Casy and Parfitt, 1986; Bagley et al., 1991; Portoghese, 1992; Bi-Yi et al., 1999; Gerak et al., 1999). Such studies can lead to an ideal analgesic with greater potency, duration of action and safety (Higashikawa & Suzuki, 2008).

We also reported on the synthesis and bioefficacy of fentanyl and its 1-substituted analogs (1–4), where the phenethyl chain of fentanyl was replaced by alkyl, ethereal and nitrile functional groups (Gupta et al., 2013). Compared to fentanyl and its four analogs, 2 exhibited the lowest median effective dose (ED50) and highest potency ratio. Therefore, with an objective to synthesize more compounds with still lower ED50 and higher potency ratio, we report here the synthesis and biological evaluation of four more 1-substituted analogs of fentanyl, viz., N-isopropyl-3-(4-(N-phenylpropionamido)piperidin-1-yl)propanamide (5), N-t-butyl-3-(4-(N-phenylpropionamido)piperidin-1-yl)propanamide (6), isopropyl 2-[4-(N-phenylpropionamido)piperidin-1-yl]propionate (7) and t-butyl 2-[4-(N-phenylpropionamido)piperidin-1-yl]propionate (8). In the present study, the phenethyl chain of fentanyl was replaced by different functional groups, viz. N-isopropyl propanamide, N-t-butyl propanamide, isopropyl propionate, and t-butyl propanoate moieties. The median lethal dose (LD50), opioid receptor-mediated activity, antinociceptive effects, ED50, and analgesic potency ratio of 5–8 were determined and compared with fentanyl as reported earlier in our publication (Gupta et al., 2013).

### Materials and methods

#### Chemistry

All the chemicals used in the present study were of the highest purity. Acrylonitrile (CAS 107-13-1), 2-bromopropionyl chloride (CAS 7148-74-5), dimethyl sulfoxide (DMSO; CAS 67-68-5) and naloxone hydrochloride (CAS 51481-60-8) were purchased from Sigma-Aldrich Inc. (St. Louis, USA). Formaldehyde (CAS: 50-00-0) was procured from Merck (Mumbai, India), isopropanol (CAS 67-63-0) was obtained from Rankem (New Delhi, India), and tert-butanol (CAS 75-65-0) was from Acros (NJ, USA).

#### Synthesis of fentanyl analogs

All the 1-substituted fentanyl analogs were synthesized from a common precursor (N-(4-piperidinyl)propionamide), prepared by the procedure reported earlier (Gupta et al., 2013), and they were found to be >98% pure. Structure and yield of fentanyl analogs are shown in Table 1.

### 4-Anilino-N-benzylpiperidine (ANBP, 2)

To a mixture of N-benzyl-4-piperidone (0.10 moles), aniline (0.10 moles) and activated zinc (0.40 moles), 10% Pd/C was added portion wise, and the mixture was stirred until the mixture was dark red. The reaction mixture was then cooled to room temperature and the mixture was poured into water (1.0 L). The organic layer was separated, and the aqueous layer was neutralized with 10% hydrochloric acid and extracted with dichloromethane (3×200 ml). The combined organic layers were washed with water (2×200 ml), brine (2×200 ml), and dried over anhydrous sodium sulfate. The solvent was removed under reduced pressure, and the residue was recrystallized from ethyl acetate and hexane to get pure N-benzyl-4-piperidone as a white solid (0.127 g, 82%).

### Table 1. Structure and yield of fentanyl analogs.

| Compound name | Substituent (R) | Molecular weight | Yield (%) | Structure |
|---------------|----------------|------------------|-----------|-----------|
| 5             | N-isopropyl-3-(4-(N-phenylpropionamido)piperidin-1-yl)propanamide | -CH2CH2NHPri | 345 | 60 | ![Structure](image1.png) |
| 6             | N-t-butyl-3-(4-(N-phenylpropionamido)piperidin-1-yl)propanamide | -CH2CH2NHBut | 359 | 65 | ![Structure](image2.png) |
| 7             | Isopropyl 2-[4-(N-phenylpropionamido)piperidin-1-yl]propionate | -CH(CH3)COOPri | 346 | 76 | ![Structure](image3.png) |
| 8             | t-butyl 2-[4-(N-phenylpropionamido)piperidin-1-yl]propionate | -CH(CH3)COOBut | 360 | 74 | ![Structure](image4.png) |

### Scheme 1. Synthetic route for fentanyl and its analogs.

- **NBP (1)**: N-Benzyl-4-piperidone
- **ANBP (2)**: 4-Anilino-N-benzylpiperidine
- **BPP (3)**: Butyl 2-(N-Benzyl-4-piperidinyl)propionate
- **PP (4)**: Propionyl 2-(N-Benzyl-4-piperidinyl)propionate

Fentanyl: R=CH2CH2Ph
5: R= CH2CH2NHPri
6: R= CH2CH2NHBut
7: R= CH(CH3)COOPri
8: R= CH(CH3)COOBut
and the resulting mixture was allowed to stir at room temperature for 24 h and at 60–70 °C in water bath for another 12 h. After completion of the reaction, the content of the flask was diluted with methanol and filtered. The filtrate was concentrated under vacuum and then neutralized with 30% ammonium hydroxide solution till pH 10. The crude product was collected by filtration and recrystallized with petroleum ether (60–80 °C) as colorless solid. Yield 85%, mp 82–83 °C. IR (KBr) νmax 3440, 3250, 3025, 2930, 2848, 1605, 1526, 1492, 1371, 1317, 1250, 1085, 975, 862, 750, 690 cm⁻¹; ¹H NMR [CDCl₃, 400 MHz]: δ=1.50 (dq, 2H, H-3ax, 5ax), 2.10 (bd, 2H, H-3eq, 5eq), 2.30 (bt, 2H, H-2ax, 6ax), 2.60 (s, 2H, H-7), 2.90 (bd, 2H, H-2eq, 6eq), 3.35 (m, 1H, H-4), 3.50 (sbr, 1H, PhN), 4.58–4.67 (m, 1H, H-4), 5.28 (m, 10 × Ar-H); ¹³C NMR [CDCl₃, 100 MHz]: δ=32.7 (CH₂, C-10), 31.8 (CH₂, C-3, 4), 46.0 (CH₂, C-7), 146.0 (CH, C-15), 138.6 (CH, C-8), 116.5 (CH, C-18), 113.0 (CH, C-16, 20), 174.8 (CO); EI-MS m/z (pos): 267 [M+1], 198 [M], 175, 146, 132, 118, 82, 77, 65, 57; Anal. Calcd. for C₁₄H₂₀N₂O: C, 72.38; H, 8.68; N, 12.06. Found: C, 72.1; H, 8.00; N, 11.99.

**N-(1-benzyl-4-piperidinyl)propionanilide (BPP; 4)**

A solution of BPP (0.07 moles) in 125 ml methanol-acetic acid mixture (3:2) was taken in a 250 ml thick-walled hydrogenation vessel containing 10% palladium on charcoal catalyst (10% w/w). The hydrogen gas was then purged into the vessel using Parr apparatus at 50 °C. When no further amount of hydrogen was consumed, the vessel was removed and the contents were filtered through celite. The filtrate was concentrated on rotary evaporator and the residue was treated with 20% aqueous sodium hydroxide solution. The aqueous solution was extracted with ethyl acetate, dried over anhydrous sodium sulphate and concentrated under reduced pressure to get the crude product. It was purified as its hydrochloride salt. Colorless crystals, yield 23.18 g (0.072 moles, 90%), mp 72–73 °C. IR (KBr) νmax 3430, 3250, 3025, 2930, 2848, 1605, 1526, 1492, 1371, 1317, 1250, 1085, 975, 862, 750, 690 cm⁻¹; ¹H NMR [CDCl₃, 400 MHz]: δ=0.94 (t, 3H, H-18), 1.25 (dq, 2H, H-3ax, 5ax), 1.55 (sbr, 1H, NH), 1.75 (bd, 2H, H-3eq, 5eq), 1.90 (q, 2H, H-10), 2.10 (bt, 2H, H-2ax, 6ax), 3.00 (bd, 2H, H-2eq, 6eq), 4.65–4.75 (m, 1H, H-4), 7.03 (d, 2 × Ar-H), 7.05 (d, 1 × Ar H), 7.35 (m, 2 × Ar-H); ¹³C NMR [CDCl₃, 100 MHz]: δ=9.5 (CH₃, C-11), 28.4 (CH₂, C-10), 31.8 (CH₂, C-3, 4), 46.0 (CH₂, C-2, 3), 52.2 (CH₂, C-14), 128.1 (CH, C-13, 17), 129.1 (CH, C-15), 130.3 (CH, C-14, 16), 138.8 (CH, C-12), 173.2 (CO); EI-MS m/z (pos): 175, 146, 132, 118, 82, 77, 68, 55, 57; Anal. Calcd. for C₁₄H₂₀N₂O: C, 72.38; H, 8.06; N, 12.06. Found: C, 72.05; H, 8.50; N, 11.99.

**Synthesis of 5 and 6**

In a three neck round bottom flask, sulfuric acid (150 mmol) was heated to 45 °C and a mixture of acrylonitrile (75 mmol) and isopropenyl/tert-butanol (74 mmol) was added drop-wise. Thereafter, the reaction mixture was stirred at 60 °C for 3 h to complete the reaction. The reaction mixture was then poured cautiously into ice cooled water with continuous stirring. The white precipitate was filtered off, washed well with plenty of water and dried under vacuum to obtain N-isopropyl/tert-butyl acrylamide. To a solution of N-isopropyl or t-butyl acrylamide (10 mmol) and N-(4-piperidinyl)propionanilide (10 mmol) in acetonitrile, silica gel (1.0 g) was added and the heterogeneous mixture was stirred at 80 °C till completion of the reaction. The reaction mixture was filtered, concentrated under vacuum and purified by flash chromatography to give the desired compounds (5 and 6).
Colorless solid (60%); mp 108–110°C. IR (KBr) ν_max 3237, 3235, 3063, 2963, 2948, 2888, 1657, 1627, 1561, 1433, 1266, 1261, 1174, 837 cm⁻¹; ¹H NMR [CDCl₃, 400 MHz]: δ=0.90 (t, 3H, H-19), 1.10 (d, 6H, H-13, 14), 1.30 (dq, 2H, H-3₃ax, 5₃ax), 1.70 (bd, 2H, H-3-eq 5-ax), 1.90 (q, 2H, H-18), 2.10 (bt, 2H, H-2-ax 6-ax), 2.40 (t, 2H, H-8), 2.60 (t, 2H, H-7), 2.90 (bd, 2H, H-2-equ 6-equ), 4.65 (m, 1H, H-4), 4.90 (m, 1H, H-12), 7.00 (d, 2 x Ar-H), 7.30 (m, 3 x Ar-H); ¹³C NMR [CDCl₃, 100 MHz]: δ=9.6 (CH₃, C-19), 22.7 (CH₃, C-13, 14), 25.4 (CH₂, C-18), 28.6 (CH₂, C-3, 5), 30.6 (CH₂, C-8), 32.4 (CH, C-12), 40.6 (CH₂, C-2, 6), 52.0 (CH, C-4), 53.4 (CH₂, C-7), 128.4 (CH, C-21, 25), 129.4 (CH, C-23), 130.3 (CH, C-22, 24), 138.8 (C, C-20), 171.4 (CO, C-16), 173.5 (CO, C-9); El-MS m/z (pos): 345, 316, 288, 245, 231, 189, 175, 159, 146, 132, 120, 93, 82, 68, 55, 44, 29; Anal. Calcd. for C₂₀H₃₀N₂O₃: C, 69.35; H, 9.04; N, 12.16. Found: C, 69.53; H, 9.04; N, 12.16.

N-t-Butyl-3-[4-(N-phenylpropionamido)piperidin-1-yl]propanamide (6)

Colorless solid (65%); mp 92–95°C. IR (KBr) ν_max 3327, 3235, 3063, 2963, 2948, 2888, 1657, 1627, 1561, 1433, 1266, 1261, 1174, 837 cm⁻¹; ¹H NMR [CDCl₃, 400 MHz]: δ=1.00 (t, 3H, H-20), 1.13 (s, 9H, H-13, 14, 15), 1.40 (dq, 2H, H-3₃ax 5₃ax), 1.70 (bd, 2H, H-3-ax 5-ax), 1.90 (q, 2H, H-19), 2.10–2.30 (m, 4H, H-2-equ 6-equ), 2.50 (t, 2H, H-7), 2.90 (bd, 2H, H-2-equ 6-equ), 4.65 (m, 1H, H-4), 7.10 (d, 2 x Ar-H), 7.40 (m, 3 x Ar-H), 8.20 (sbr, 1H, NH); ¹³C NMR [CDCl₃, 100 MHz]: δ=9.6 (CH₃, C-19), 14.8 (CH₃, C-8), 21.8 (CH₂, C-13, 14), 22.0 (CH₂, C-18), 28.5 (CH₂, C-3, 5), 47.4 (CH₂, C-2, 6), 50.9 (CH, C-4), 62.4 (CH, C-7), 68.4 (CH, C-12), 128.4 (CH, C-21, 25), 129.4 (CH, C-23), 130.3 (CH, C-22, 24), 138.7 (C, C-20), 171.6 (CO, C-16), 173.6 (CO, C-9); El-MS m/z (pos): 346, 331, 317, 289, 259, 216, 203, 187, 146, 132, 110, 93, 82, 56; Anal. Calcd. for C₂₀H₂₃N₃O₂: C, 69.35; H, 8.67; N, 8.05. Found: C, 69.33; H, 8.73; N, 8.09.

### Synthesis of 7 and 8

To a solution of Isopropanol/tert-butanol (38 mmol) in toluene (25 ml), 2-bromopropionyl chloride (38 mmol) was added drop-wise and the reaction mixture was stirred at room temperature for 4 h under nitrogen atmosphere. The reaction mixture was quenched with water and extracted with ethyl acetate. The organic phase was washed with saturated sodium bicarbonate solution and brine, dried over sodium sulfate and concentrated under reduced pressure to get the desired ester as oils. Thereafter, to a solution of isopropyl or t-butyln 2-bromopropionate (10 mmol) and N-(4-piperidinyl)propionamide (10 mmol) in DMF-water, potassium carbonate was added and the reaction mixture was stirred at 80°C. When the reaction was complete, the reaction mixture was filtered, and the filtrate was diluted with water and extracted with ethyl acetate. The organic phase was separated, and the aqueous phase was extracted with ethyl acetate; the combined organic phase was washed with brine, dried over sodium sulfate and concentrated under vacuum to give the crude product which was purified by flash chromatography to give the desired compounds (7 and 8).
Colorless solid (74%); mp 98–100 °C. IR (KBr) v_max 3330, 3030, 2984, 2880, 1740, 1652, 1470, 1113, 1006, 929, 801; 1H NMR (CDCl_3, 400 MHz): δ = 1.00 (t, 3H, H-20), 1.18 (d, 3H, H-8), 1.34 (m, 2H, H-3ax, 5ax), 1.45 (s, 9H, H-13, 14, 15), 1.75 (m, 2H, H-3eq, 5eq), 1.90 (q, 2H, H-19), 2.35–2.47 (m, 2H, H-2ax, 6ax), 2.90 (bd, 2H, H-2eq, 6eq), 3.10 (q, 3H, H-7), 4.66 (m, 1H, H-4), 7.07 (d, 2 × Ar-H), 7.38 (m, 3 × Ar-H); 13C NMR (CDCl_3, 100 MHz): δ = 9.6 (CH_3, 1H, H-7), 4.66 (m, 1H, H-4), 7.07 (d, 2 × Ar-H), 7.38 (m, 3 × Ar-H); 1H NMR (CDCl_3, 400 MHz): δ = 1.00 (t, 3H, H-20), 1.18 (d, 3H, H-8), 1.34 (m, 2H, H-3ax, 5ax), 1.45 (s, 9H, H-13, 14, 15), 1.75 (m, 2H, H-3eq, 5eq), 1.90 (q, 2H, H-19), 2.35–2.47 (m, 2H, H-2ax, 6ax), 2.90 (bd, 2H, H-2eq, 6eq), 3.10 (q, 3H, H-7), 4.66 (m, 1H, H-4), 7.07 (d, 2 × Ar-H), 7.38 (m, 3 × Ar-H).

**Biological assay**

**Animals**

Male Swiss albino mice (25–30 g) were procured from the Animal Facility of the Defence Research and Development Establishment (DRDE), Gwalior. The animals were housed in polypropylene cages on dust free rice husk as bedding material, with free access to food (Ashirwad Brand, Chandigarh, India) and water ad libitum. Prior to experiment, the animals were randomly and acclimatized for seven days in controlled environmental conditions (22±2 °C; relative humidity 40–60%) at a 12 h light/12 h dark cycle. The care and maintenance of the animals were as per the approved guidelines of the Committee for the Purpose of Control and Supervision of Experiments on Animals (CPCSEA), Ministry of Environment and Forest, Govt. of India, New Delhi, India. The experimental protocol was approved by the Institutional Ethical Committee on Animal Experiments approved by CPCSEA.

**Determination of LD_{50}**

The LD_{50} of the compounds was determined by intravenous (i.v.) and oral (p.o.) routes following Dixon’s up and down method (Dixon, 1965), using 4–6 mice for each value. All the compounds were dissolved in DMSO and administered in a volume <10 ml/kg body weight. To determine the i.v. LD_{50} the compounds were administered through the tail vein using a 27 gauge needle, and for the p.o. LD_{50} a 16 gauge oral feeding cannula (HSE-Harvard, Germany) was used. Although, the animals were observed for 14 days, the mortality invariably occurred within the first 24 h. All the dead animals were autopsied to see any visceral changes. All the observations were compared with those of fentanyl (Gupta et al., 2013).

**Observational assessment**

Observational assessment on spontaneous activities of the CNS, peripheral nervous system (PNS) and autonomic nervous system (ANS) was performed as per the modified method discussed elsewhere (Irwin, 1964, 1968). Mice were divided into five groups of twenty seven animals each as follows: (i) vehicle control (DMSO), (ii) 5, (iii) 6, (iv) 7 and (v) 8. Three animals from each group received 0.25, 0.50, and 0.75 LD_{50} each of the compounds by i.v., i.p., and p.o. routes. Immediately after treatment, the animals were closely observed for 2 h by a blind observer for various CNS, PNS and ANS activities (Gupta et al., 2013).

**Determination of opioid antagonist activity**

To confirm the opioid receptor-mediated effects of the compounds, mice were treated with 0.50 LD_{50} (i.p.) of the analogs (5–8), in the presence or absence of naloxone hydrochloride (5.0 or 10.0 mg/kg; s.c.; –10 min). Soon after, the animals were closely observed for various neurotoxic manifestations of the compounds and their disappearance in the presence of naloxone hydrochloride (Matosiuk et al., 2002; Leavitt, 2009). Three animals were used for each treatment.

**Determination of ED_{50} potency ratio and therapeutic index**

Analogical ED_{50} of all the compounds was determined by formalin-induced hind paw licking method (Hunskar & Hole, 1987). The compounds were dissolved in DMSO and administered (i.p.) 30 min prior to administration of formalin (2.5%, 20 μl) injected sub-plantarly in one hind paw. The duration of paw licking as index of nociception was monitored at 0–5 min (first phase or neurogenic phase) and 15–30 min (second phase or inflammatory phase). Each compound was evaluated at four different doses, using four animals for each dose. Thereafter, Litchfield and Wilcoxon (1949) method was utilized for statistical evaluation of data and calculation of ED_{50} values. The potency ratio was determined as the ratio of ED_{50} of fentanyl and the ED_{50} of each analogue. The therapeutic index was calculated as the ratio of LD_{50} of the analogs and their corresponding ED_{50}. All the values were compared with those of fentanyl (Gupta et al., 2013).

**Measurement of analgesic activity**

Analogical activity of the analogs was assessed at their ED_{50} by formalin-induced hind paw licking test (Hunskar & Hole, 1987) and tail immersion test (Janssen et al., 1963). To perform the hind paw licking test, mice were divided into six groups of six animals each and given various treatments (i.p.) as follows: (i) vehicle control (DMSO), (ii) fentanyl (28.0 μg/kg), (iii) 5 (20.5 μg/kg), (iv) 6 (21.0 μg/kg), (v) 7 (35.5 μg/kg), and (vi) 8 (55.0 μg/kg). Thirty minutes after administration of different compounds, formalin (2.5%, 20 μl) was injected sub-plantarly in one hind paw. The duration of paw licking as index of nociception was monitored at 0–5 min (neurogenic phase) and 15–30 min (inflammatory phase). To conduct the tail immersion test, mice received various treatments as discussed above. The distal part (2–3 cm) of the tail was immersed in hot water maintained at 55.0±1.0 °C, and the time taken by the mice to deflect or withdraw the tail was recorded as the reaction time. A cut off time of tail immersion was taken as 10 sec, and thereafter the measurement was stopped to avoid any tissue injury. The initial reading was taken.
immediately before treatment and then at 15, 30, 45, and 60 min post treatment. Tail withdrawal in vehicle treated mice usually occurs between 2.6 and 3.0 sec. Therefore, a withdrawal time of >3 sec was considered as a positive response. Prior to the analgesic test, the animals were screened by immersing their tail in hot water (55.0±1.0 °C) and only those animals were selected for the experiment which showed tail withdrawal latency of <5 sec.

**Statistics**

Results of analgesic activity were expressed as mean ± SE (n=6). The statistical analysis was carried out by Student’s t test, using SigmaStat software (SSP Inc., USA). Statistical significance was drawn at p<0.05 and p<0.01.

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**Table 2.** LD$_{50}$ of fentanyl analogs by different routes in mice.

| Comp. | Intravenous | Intraperitoneal | Oral |
|-------|-------------|----------------|------|
| 5     | 57.0 (42.1–77.2) | 113.8 (84.3–153.9) | 285.8 (211.2–386.7) |
| 6     | 45.3 (33.5–61.3) | 107.3 (81.5–141.3) | 220.7 (163.2–298.7) |
| 7     | 35.0 (25.9–47.3) | 277.9 (205.4–376.0) | 717.9 (530.6–971.4) |
| 8     | 44.0 (32.6–59.6) | 349.0 (258.6–473.4) | 903.9 (668.0–1223.0) |

Fentanyl analogs, viz., 5-8 were administered through different routes and acute (24 h) LD$_{50}$ was determined by Dixon’s up and down method (Dixon, 1965). Values in parentheses are fiducial limits at 95% confidence interval.

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**Table 3.** Observational assessment after intravenous administration of fentanyl analogs in mice.

| Comp. | Dose | CNS | PNS | Reflexes | ANS |
|-------|------|-----|-----|----------|-----|
|       | –    | 0   | 0   | 0 0 0 0 0 |
| Low   | + +  | + + | + + |          |     |
| Medium | +++ | +++ | ++ | + + + + + |
| High  | +++ | +++ | +++ | +++ + + + |
| Low   | + + + | + + + | + + + + + |
| Medium | ++ | ++ | + + | + + + + + |
| High  | +++ | +++ | +++ | +++ + + + |
| Low   | ++ | ++ | + + | + + + ++ |
| Medium | +++ | +++ | ++ | + + + + + |
| High  | +++ | +++ | ++ | + + + + + |

Mice were intravenously administered 0.25 (Low), 0.50 (Medium), and 0.75 (High) LD$_{50}$ of fentanyl analogs, viz., 5-8. The control animals received DMSO. Immediately after treatment, the animals were closely observed for 2 h by a blind observer for its effects on CNS, PNS (effects after manipulation and effects on reflexes) and ANS activities. The scorings were given as: 0 (no observational change), + (little activity), ++ (moderate flexibility), +++ (strong response) and ++++ (exaggerated response). Each treatment included three animals.

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**Table 4.** Observational assessment after intraperitoneal administration of fentanyl analogs in mice.

| Comp. | Dose | CNS | PNS | Reflexes | ANS |
|-------|------|-----|-----|----------|-----|
|       | –    | 0   | 0   | 0 0 0 0 0 |
| Low   | + +  | + + | + + |          |     |
| Medium | +++ | +++ | ++ | + + + + + |
| High  | +++ | +++ | +++ | +++ + + + |
| Low   | + + + | + + + | + + + + + |
| Medium | ++ | ++ | + + | + + + + + |
| High  | +++ | +++ | +++ | +++ + + + |
| Low   | ++ | ++ | + + | + + + ++ |
| Medium | +++ | +++ | ++ | + + + + + |
| High  | +++ | +++ | ++ | + + + + + |

Mice were intraperitoneally administered 0.25 (Low), 0.50 (Medium), and 0.75 (High) LD$_{50}$ of fentanyl analogs, viz., 5-8. The control animals received DMSO. Immediately after treatment, the animals were closely observed for 2 h by a blind observer for its effects on CNS, PNS (effects after manipulation and effects on reflexes) and ANS activities. The scorings were given as: 0 (no observational change), + (little activity), ++ (moderate flexibility), +++ (strong response) and ++++ (exaggerated response). Each treatment included three animals.

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**Results**

**Toxicity of fentanyl analogs**

Table 2 shows the LD$_{50}$ values of fentanyl analogs by i.v., i.p. and p.o. routes in mice. All the analogs were found to be less toxic compared to fentanyl by all the routes of administration. The LD$_{50}$ of fentanyl was 6.9, 17.5 and 27.8 mg/kg by i.v., i.p., and p.o. routes, respectively (Gupta et al., 2013). On the basis of LD$_{50}$, the order of toxicity of the compounds was: fentanyl >7 >8 >6 >5 for i.v. route, and fentanyl >6 >5 >7 >8 for i.p. and p.o. routes. On the basis of LD$_{50}$ by i.v. route, all the analogs showed more or less similar toxicity, but by i.p. and p.o. route, 7 and 8 were distinctly less toxic compared to 5 and 6. Mice succumbing to lethal doses of the compounds were autopsied immediately, which revealed profuse intestinal hemorrhage.

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**Observational assessment**

Observational assessment based on CNS, PNS and ANS activities was made following administration of three doses of the compounds by i.v., i.p. and p.o. routes. The CNS activities included spontaneous motor activity, restlessness, grooming behavior, squatting, staggering, ataxic gait, lying flat on the belly, lying flat on the side, lying flat on the back, sleeping, narcosis, bizarre behavior, timidity, Straub’s phenomenon, writhing, tremors, twitches, opisthotonus, clonic convulsions, tonic convulsions, rolling and jumping and convulsions. The PNS activities (after...
Table 5. Observational assessment after oral administration of fentanyl analogs in mice.

| Comp. | Dose | CNS | After manipulations | Reflexes | ANS |
|-------|------|-----|---------------------|---------|-----|
| Control | –    | 0   | 0                   | 0       | 0   |
| 5     | Low  | +   | 0                   | 0       | 0   |
|       | Medium | ++ | 0                   | +       | 0   |
|       | High  | +++| +++                 | +       | +   |
| 6     | Low  | +   | 0                   | 0       | 0   |
|       | Medium | ++ | 0                   | +       | 0   |
|       | High  | +++| +++                 | +       | +   |
| 7     | Low  | +   | 0                   | 0       | 0   |
|       | Medium | ++ | 0                   | +       | 0   |
|       | High  | +++| +++                 | +       | +   |
| 8     | Low  | +   | 0                   | 0       | 0   |
|       | Medium | ++ | 0                   | +       | 0   |
|       | High  | +++| +++                 | +       | +   |

Mice were orally administered 0.25 (Low), 0.50 (Medium), and 0.75 (High) LD50 of fentanyl analogs, viz. 5–8. The control animals received DMSO. Immediately after treatment, the animals were closely observed for 2 h by a blind observer for its effects on CNS, PNS (effects after manipulation and effects on reflexes) and ANS activities. The scorings were given as: 0 (no observational change), + (little activity), ++ (moderate flexibility), +++ (strong response) and ++++ (exaggerated response). Each treatment included three animals.

Figure 1. Analgesic activity of fentanyl and its four analogs, viz. 5–8, was measured by formalin-induced hind paw licking method in mice. Values are expressed as mean ± SE (n=6). *p<0.01 (Student’s t-test).

Table 6. ED50 potency ratio and therapeutic index of fentanyl analogs in mice.

| Comp. | ED50 (μg/kg) | Potency ratio | Therapeutic index |
|-------|--------------|---------------|-------------------|
| 5     | 20.5 (11.3–37.1) | 1.37 (1.32–1.41) | 5551.2 (4148.2–7442.5) |
| 6     | 21.0 (11.4–38.6) | 1.33 (1.31–1.36) | 5109.5 (3660.6–7149.1) |
| 7     | 35.5 (17.4–72.4) | 0.79 (0.86–0.72) | 7828.2 (5193.4–11804.6) |
| 8     | 55.0 (32.9–91.9) | 0.51 (0.45–0.57) | 6361.8 (5151.3–7860.2) |

Analgesic ED50 of fentanyl analogs, viz. 5–8 was determined by formalin-induced hind paw licking method (Hunskar & Hole, 1987; Litchfield & Wilcox, 1949). The potency ratio was determined as the ratio of ED50 of fentanyl 28.0 (μg/kg) (Gupta et al., 2013) and ED50 of each analog. The therapeutic index was calculated as the ratio of LD50 of the compounds and their ED50. Values in parentheses are fiducial limits at 95% confidence interval.

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Bioassay of fentanyl analogs
Shiv Kumar Yadav, Chandra Kant Maurya, Pradeep Kumar Gupta, Ajai Kumar Jain, Kumaran Ganesan, Rahul Bhattacharya

Discussion

Although fentanyl is widely used as a narcotic analgesic agent, it has been implicated in drug abuse and fatalities due to overdosing and narrow therapeutic window (Yassen et al., 2008; Jumbelic, 2010). Over many recent years, SAR and molecular modeling of several new analogs of fentanyl have projected many potent compounds (Vučković et al., 2000). Our recent work also revealed some effective 1-substituted analogues of fentanyl (Gupta et al., 2013). The present study reports the synthesis, and biological evaluation of four new fentanyl analogs. The LD50 of the new analogs by different routes revealed that all the compounds were less toxic compared to fentanyl, thus showing an improved margin of safety over fentanyl. Autopsy of the animals succumbing to high doses of the compounds showed severe intestinal hemorrhage. This possibly occurred due to pooling of blood following hypovolemic shock. Similar observations were also made during our previous study (Gupta et al., 2013) and after administration of methyl-substituted and para-substituted fentanyl analogs (Higashikawa & Suzuki, 2008).

Observational assessment on spontaneous CNS, PNS, and ANS activities is usually performed to evaluate the psychotropic activity and toxicity of the compounds (Irwin, 1964, 1968). Observational assessment made in the present study revealed that all the compounds exerted significant dose-dependent influence on CNS and PNS activities. Also, the compounds were found to induce straub’s phenomenon, catalepsy, rigidity, circling and stereotypical behavior, which are distinctive characteristics of opioid analgesics. Severe convulsions, a typical attribute of morphine intoxication, were also observed. This can be attributed to inhibition of release of gamma-aminobutyric acid by interneurons (McGinty & Friedman, 1988). All the compounds were evaluated through parenteral and p.o. routes of administration, which are the preferred routes of opioids for pain management (Gardocki & Yelnosky, 1964; Hallenbeck, 2003; Vučkovic et al., 2011). In the present study, most of the observations on motor coordination and behavioral tests of fentanyl analogs were very similar to previous observations with fentanyl (Gardocki & Yelnosky, 1964). There were also reduced ANS activities, like defecation and micturation, which are typical of opioid analgesics (Gupta et al., 2013).

Short acting opioid antagonists, such as naloxone, have been successfully used to rapidly reverse the neurotoxic effects of opioid overdose (Leavitt, 2009). Naloxone is a nonselective antagonist of opioid receptors, and is generally used to verify any opioid-mediated effects of the drugs.

Figure 2. Analgesic activity of fentanyl and its four analogs, viz. 5-8, was measured by tail immersion test. Values are expressed as mean ± SE (n=6). *p< 0.05 and **p< 0.01 (Student’s t test).

formalin-induced hind paw licking method. The duration of paw licking as index of nociception was observed at the first phase (0–5 min) and second phase (15–30 min). In the first phase, none of the compounds displayed any analgesic activity because the time spent in paw licking did not decrease compared to control. However, in the second phase, all the compounds exhibited significant (p<0.01) analgesic activity compared to control. Further, compared to control, the percentage of decrease in time of paw licking was 78.5, 68.8, 69.2, 44.3, and 49.9 for fentanyl, 5, 6, 7, and 8, respectively. The data show that in the second phase, antinociceptive activity of 7 and 8 was lower than that of other compounds. Figure 2 refers to the analgesic activity of fentanyl and its analogs measured by tail immersion method. Animals receiving ED50 of the compounds were observed to increase the reaction time at this time also both 5 and 6 showed significant analgesic activity at 15, 30 and 45 min post-treatment, the maximum being at 30 min. The analgesic activity of 5 and 6 was evident at all the time points. The analgesic activity of 5 progressively increased after 15 min to a maximum at 45 min, thereafter it declined at 60 min. On the other hand, analgesic activity of 6 was maximum at 15 min, which gradually declined by 60 min. However, at this time also both 5 and 6 were significantly (p<0.05) different from the corresponding control. Compound 7 and 8 displayed significant analgesic activity at 15 and 30 min only, the maximum for both being observed at 30 min. In brief, both 5 and 6 demonstrated longer antinociceptive activity compared to fentanyl, while the effects of 7 and 8 were short-lived.

Tail withdrawal latency (sec)
In the present study, the analgesic activity of 1-substituted analogs of fentanyl was determined by formalin-induced hind paw licking method (Hunskaar & Hole, 1987) and tail immersion test (Janssen et al., 1963). The formalin test is a widely used model for screening novel compounds for the treatment of neuropathic pain. The method involves a behavioral nociceptive test that assesses the response of the animal to moderate and continuous pain (Meunier et al., 1998). Formalin produces biphasic pain behavior. The first phase (i.e. neurogenic phase) is due to the direct effect of formalin on nociceptors, while the second phase (i.e. inflammatory phase) is due to the development of an inflammatory response caused by tissue injury leading to the release of histamine, serotonin, prostaglandin and excitatory amino acids (Correa & Calixto, 1993; Damas & Liegeois, 1999). In the present study, all the analogs were found to be more effective in the second phase, which could be due to their implications as inhibitors of pain mediators during the late phase. In the present study, 5 and 6 exhibited higher potency compared to fentanyl but lower analgesic activity when evaluated at respective ED50. Potency and efficacy are different concepts, and when an agonist possesses high potency, it need not display also high efficacy, and vice versa (Lambert, 2004). An agonist capable of producing the maximum response in that system is termed a full agonist and anything that produces a lower response is a partial agonist. The ability of the agonist to bind to the receptor will determine the ability to produce a response and to some extent the size of that response (Lambert, 2004). The tail immersion test is widely employed for opioid analgesics. This method gives intensity, onset, peak, duration of action and safety of fentanyl and other morphine-like behavioral effects in rats (Vukovic et al., 2012).

In the present study, the analgesic activity of 1-substituted analogs of fentanyl was determined by formalin-induced hind paw licking method (Hunskaar & Hole, 1987) and tail immersion test (Janssen et al., 1963). The formalin test is a widely used model for screening novel compounds for the treatment of neuropathic pain. The method involves a behavioral nociceptive test that assesses the response of the animal to moderate and continuous pain (Meunier et al., 1998). Formalin produces biphasic pain behavior. The first phase (i.e. neurogenic phase) is due to the direct effect of formalin on nociceptors, while the second phase (i.e. inflammatory phase) is due to the development of an inflammatory response caused by tissue injury leading to the release of histamine, serotonin, prostaglandin and excitatory amino acids (Correa & Calixto, 1993; Damas & Liegeois, 1999). In the present study, all the analogs were found to be more effective in the second phase, which could be due to their implications as inhibitors of pain mediators during the late phase. In the present study, 5 and 6 exhibited higher potency compared to fentanyl but lower analgesic activity when evaluated at respective ED50. Potency and efficacy are different concepts, and when an agonist possesses high potency, it need not display also high efficacy, and vice versa (Lambert, 2004). An agonist capable of producing the maximum response in that system is termed a full agonist and anything that produces a lower response is a partial agonist. The ability of the agonist to bind to the receptor will determine the ability to produce a response and to some extent the size of that response (Lambert, 2004). The tail immersion test is widely employed for opioid analgesics. This method gives intensity, onset, peak, duration of action and safety of fentanyl and other morphine-like behavioral effects in rats (Janssen et al., 1963). In the present study, onset, peak and duration of the analgesic effect of all the analogs were compared with those of fentanyl by using tail immersion test. In order to perform this study, all the compounds were tested at their ED50. We found that 5 and 6 produced analgesia for a longer duration compared to fentanyl. Most of the opioid analogs exert their analgesic and adverse effects primarily through MOR. However, individual strong opioids may interact, at least in part, with different opioid receptor sub-populations or modulate MOR signaling in different ways (Pasternak, 2004; Lee et al., 2007), which may improve tolerability (Ananthan, 2006; Smith, 2008; Spetea et al., 2010). Previous studies have shown that 4-methyl fentanyl was four times more potent than fentanyl (Micovic et al., 2000), while introduction of 3-carbomethoxy group in the piperidine ring of fentanyl reduced the potency but did not affect the tolerability and safety (Vučković et al., 2011). The shorter duration of action of 3-carbomethoxy fentanyl in comparison with fentanyl might be due to the susceptibility of the carbomethoxy group to rapid hydrolysis by non-specific esterases (Feldman et al., 1991). It is also conceivable that the introduction of a 3-carbomethoxy group in the piperidine ring affects the duration of action by altering physicochemical properties (Scholz et al., 1996).

Conclusion

Opioid analgesics are usually prescribed for acute and chronic pain management but their use is restricted due to undesirable side effects and a narrow therapeutic window. The present study addresses the synthesis and biological evaluation of four new 1-substituted analogs of fentanyl, where the phenethyl tail of fentanyl was replaced by different functional groups. All the compounds exhibited 8–12 fold increase in therapeutic index but only two compounds (5 and 6) produced lower ED50 and higher potency ratio compared to fentanyl. Thus out of the four compounds tested, only two were found to be promising for further studies on pain management.

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