Phase Transition and Electronic Structures of All-\textit{d}-Metal Heusler-Type $X_2\text{MnTi}$ Compounds ($X = \text{Pd}, \text{Pt}, \text{Ag}, \text{Au}, \text{Cu}, \text{and Ni}$)

Mengxin Wu$^1$, Feng Zhou$^1$, Rabah Khenata$^2$, Minquan Kuang$^1$* and Xiaotian Wang$^1$*

$^1$ School of Physical Science and Technology, Southwest University, Chongqing, China. $^2$ Laboratoire de Physique Quantique de la Matérie et de la Modélisation Mathématique (LPQ3M), Université de Mascara, Mascara, Algeria

In this work, we investigated the phase transition and electronic structures of some newly designed all-\textit{d}-metal Heusler compounds, $X_2\text{MnTi}$ ($X = \text{Pd}, \text{Pt}, \text{Ag}, \text{Au}, \text{Cu}, \text{and Ni}$), by means of the first principles. The competition between the $XA$ and $L2_1$ structures of these materials was studied, and we found that $X_2\text{MnTi}$ favors to feature the $L2_1$-type structure, which is consistent with the well-known site-preference rule (SPR). Under the $L2_1$ structure, we have studied the most stable magnetic state of these materials, and we found that the ferromagnetic state is the most stable due to its lower energy. Through tetragonal deformation, we found that the $L2_1$ structure is no longer the most stable structure, and a more stable tetragonal $L1_0$ structure appeared. That is, under the tetragonal strain, the material enjoys a tetragonal phase transformation (i.e., from cubic $L2_1$ to tetragonal $L1_0$ structure). This mechanism of $L2_1$-$L1_0$ structure transition is discussed in detail based on the calculated density of states. Moreover, we found that the energy difference between the most stable phases of $L1_0$ and $L2_1$, defined as $\Delta E_M (\Delta E_M = E_{\text{Cubic}} - E_{\text{Tetragonal}})$, can be adjusted by the uniform strain. Finally, the phonon spectra of all tetragonal $X_2\text{MnTi}$ ($X = \text{Pd}, \text{Pt}, \text{Ag}, \text{Au}, \text{Cu}, \text{and Ni}$) phases are exhibited, which provides a powerful evidence for the stability of the tetragonal $L1_0$ state. We hope that our research can provide a theoretical guidance for future experimental investigations.

Keywords: spintronic, electronic structure, DFT, electronic properties, Heusler alloys

INTRODUCTION

Magnetic shape memory compounds (MSMAs) (O’Handley, 1998) are a new type of intelligent materials which integrates magnetic controlled shape memory and magnetic field-induced strain simultaneously. It can be used as key components of sensors and brakes in the future. MSMAs have both thermoplastic martensitic transformation (Oikawa et al., 2001) and magnetic transformation (Oikawa et al., 2002), and their shape memory effect can be controlled by the magnetic field. That is to say, under the effect of magnetic field, their size or volume will be changed, resulting in a great strain, i.e., the magnetostriective effect (Populoh et al., 2012). In addition, MSMAs enjoy magnetoresistance (Ullakko et al., 1996) and magnetocaloric (GschneidnerJr et al., 2005) effects, so they have been regarded as a research hot spot in recent years.

Heusler (Graf et al., 2011; Birkel et al., 2013; Ahmadian and Salary, 2014; Kirievsky et al., 2014; Xue et al., 2016; Miranda and Gruhn, 2017; Ghunaim et al., 2018; Li et al., 2019, 2020) compounds
belong to intermetallic compounds. Heusler compounds naturally have many excellent properties, such as high Curie temperature \( (T_C) \) (Wurmehl et al., 2005), adjustable electronic structure, suitable semiconductor lattice constants, and various magnetic properties. Therefore, Heusler compounds can be seen as good candidates for spin gapless semiconductors (SGSs) (Gao and Yao, 2013; Skafitowrso et al., 2013; Wang et al., 2016; Wang X. et al., 2017), thermoelectric materials (Downie et al., 2013; Huang et al., 2018; Mallick and Gupta, 2019; Hao et al., 2020), and topological insulators (Hou et al., 2015; Lin et al., 2015). Because Heusler compounds possess excellent properties, they have been regarded as a research hot spot over the past 100 years. To date, researchers have discovered thousands of Heusler compounds. Heusler compounds roughly can be divided into three structures, namely, full Heusler (Wang X. T. et al., 2017), half Heusler (Silpawilawan et al., 2017), and quaternary Heusler (Cui et al., 2019), whose stoichiometric compositions are \( X_2YZ \), \( XXY \), and \( XYMZ \), respectively. X, Y, and M are usually transition elements, while the Z atom is a main group element. In recent years, a new type of Heusler compounds has been found by researchers, and this type of Heusler compounds is named as all-\( d \)-metal Heusler compounds (Wei et al., 2015).

In the early 1990s, all-\( d \)-metal Heusler compounds \( \text{Zn}_2\text{AuAg} \) and \( \text{Zn}_2\text{CuAu} \) (Muldawer, 1966) were synthesized experimentally, which proved that all-\( d \)-metal Heusler compounds can be successfully prepared. This interesting study opened up a new direction for the research of Heusler compounds. All-\( d \)-metal Heusler compounds are different from common Heusler compounds in that they are composed of transition metal elements without the participation of main group elements. Compared with common Heusler compounds, all-\( d \)-metal Heusler compounds have the following advantages: (1) all-\( d \)-metal Heusler compounds do not have many restrictions on atomic site preference, so they can show more phase space and versatility than traditional materials; (2) they have high strength and toughness; and (3) in addition to magnetic phase transitions, all-\( d \)-metal Heusler compounds also have many untouched physical properties, such as spintronics.

Recently, an effective method, i.e., adjusting composition, was proposed by Tan et al. (2019) to make regular the phase transition of all-\( d \)-metal Heusler compounds. In their work, the atomic ordering, structural stability, tetragonal deformation, magnetism, and electronic structures of the Mn–Ni–V system, including \( \text{Mn}_2-x\text{Ni}_{1.5+x}\text{V}_{0.5} \), \( \text{Mn}_2-x\text{Ni}_{1+x}\text{V} \), and \( \text{Mn}_2-x\text{Ni}_{0.5+x}\text{V}_{1.5} \) compounds, are studied by employing the first principles. They stated that the tetragonal phase is more stable than the cubic phase for these all-\( d \)-metal Heusler compounds: \( \text{MnNi}_2\text{V} \), \( \text{Mn}_{1.25}\text{Ni}_{1.75}\text{V} \), \( \text{MnNi}_2\text{S}_{0.5}\text{V}_{0.5} \), and \( \text{Mn}_{0.5}\text{Ni}_{4}\text{V}_{1.5} \).

Based on the above information, we focus on the electronic structures and phase transition of all-\( d \)-metal Heusler compounds \( X_2\text{MnTi} \) (\( X = \text{Pd}, \text{Pt}, \text{Ag}, \text{Au}, \text{Cu}, \text{and Ni} \)) with the help of the first principles. To the best of knowledge, the physical properties of the \( X_2\text{MnTi} \) system have not been studied yet by other researchers up to now. We will reflect that a \( L2_1-L1_0 \) phase transition can be found in all-\( d \)-metal Heusler compounds \( X_2\text{MnTi} \) (\( X = \text{Pd}, \text{Pt}, \text{Ag}, \text{Au}, \text{Cu}, \text{and Ni} \)) under the tetragonal distortion. The effect of uniform strain on the cubic–tetragonal transition for this \( X_2\text{MnTi} \) system was also discussed in detail in this work. The mechanism of the \( L2_1-L1_0 \) phase transition is discussed according to the calculated density of states in both cubic and tetragonal phases, and finally, we will further prove the stability of the tetragonal phase by calculating the phonon spectrum of the tetragonal \( L1_0 \) phase.

### COMPUTATIONAL METHODS

Electronic structure calculations were performed using density functional theory (DFT), within the VASP code (Hafner, 2007). The exchange–correlation potential is treated by using the generalized gradient approximation (GGA) (Perdew et al., 1996) with the Perdew–Burke–Ernzerhof (PBE) (Perdew et al., 1998) functional. We also use the projection enhanced wave (PAW) (Kresse and Furthmüller, 1996) method to deal with the interaction between the ion nucleus and valence electron. In the calculations, the cutoff energy was set at 450 eV. A Monkhorst-Pack special 12 \( \times \) 12 \( \times \) 12 k-point mesh was used in the Brillouin zone (BZ) integration. The unit cell was optimized until the force and total energy were \( < 0.001 \text{ eV/Å} \) and \( 1 \times 10^{-5} \text{ eV} \), respectively. The crystal models of cubic \( L2_1 \) phase and tetragonal \( L1_0 \) phase of Heusler alloys are built via VESTA. To calculate the dynamical stabilities of these alloys, phonon dispersion is obtained by means of the force-constants method using Nanodcal code. For the \( L2_1 \)-type \( X_2\text{MnTi} \) alloys, two magnetic states are considered: one is the ferromagnetic state with spin orderings of X-1, X-2, Mn, and Ti that are \( \uparrow \uparrow \uparrow \uparrow \); the other one is the antiferromagnetic state with spin orderings of X-1, X-2, Mn, and Ti that are \( \downarrow \downarrow \downarrow \downarrow \), respectively.

### RESULTS AND DISCUSSION

#### The Crystal and Magnetic Structures of the \( X_2\text{MnTi} \) (\( X = \text{Pd}, \text{Pt}, \text{Ag}, \text{Au}, \text{Cu}, \text{and Ni} \)) Heusler Compounds

Heusler compounds, \( X_2YZ \), enjoy a highly ordered cubic structure (Han et al., 2019a). There are generally four different positions in a primitive cell, namely, A (0, 0, 0), B (0.25, 0.25, 0.25), C (0.5, 0.5, 0.5), and D (0.75, 0.75, 0.75), respectively. The transition metal elements X and Y occupy the A, B, and C positions, and the main group element Z is preferred to occupy the D position. Different occupation positions of X and Y atoms will lead to different structures, namely, \( L2_1 \) and \( XA \) structures (Suzuki and Kyono, 2004) (as shown in Figures 1A,B). In the former, two X atoms occupy A and C positions, and Y and Z atoms enter B and D positions, respectively; in the latter, two X atoms occupy A and B positions, and Y and Z atoms are located at C and D positions, respectively.

Next, we studied the competition between the \( XA \) structure and the \( L2_1 \) structure of \( X_2\text{MnTi} \). In Figure 2, the total energy...
of X$_2$MnTi at the ground state with different structures (L$_2^1$ and XA) is determined, and we set the ground state energy of the L$_2^1$ structure as 0 eV as reference. One can clearly see that, for X$_2$MnTi, the total energy of XA is higher than L$_2^1$, reflecting that the most stable ordered structure is L$_2^1$ for the X$_2$MnTi system. In Figure 3, we further give a comparison of the total energy of two magnetic states [ferromagnetic (FM) and antiferromagnetic (AFM)] for the L$_2^1$-type X$_2$MnTi. We set the total energy of AFM to 0 eV as reference. As shown in Figure 3, the energy of X$_2$MnTi in the FM structure is lower than 0, which indicates that the X$_2$MnTi tends to exhibit the FM magnetic state.

### The L$_2^1$-L$_{10}$ Phase Transition of All-d-Metal Heusler-Type X$_2$MnTi

In this section, our research goal is to explore the possible competition between the L$_2^1$ (see Figure 4A) and L$_{10}$ (see Figure 4B) of the all-d-metal Heusler compounds X$_2$MnTi. We used Bain paths to investigate the reversible transformation between the ordered L$_2^1$ and L$_{10}$ phases during tetragonal distortion. Bain paths (Alippi et al., 1997) have tetragonal states along the geometries which connect the bcc and fcc phases of a material. It is assumed that there is no cell volume change after applying deformation on the cubic phase, then the energy

![Crystal structure of the (A) XA-type X$_2$MnTi and (B) L$_2^1$-type X$_2$MnTi (X = Pd, Pt, Ag, Au, Cu, and Ni). X is highlighted by gray balls, Ti is marked by blue balls, and Mn is marked by purple balls, respectively.](image1)

![Total energy difference E(XA) - E(L$_2^1$) of X$_2$MnTi under the L$_2^1$ and XA structure (in FM state) with the energy of L$_2^1$ set as 0 eV for reference.](image2)
The difference $\Delta E$ in respect to $c/a$ ratios is calculated to estimate whether it is a stable or metastable phase. It is a commonly used method to predict reversible transformation, and a number of literature (Barman et al., 2007; Özdemir Kart et al., 2008; Qawasmeh and Bothina, 2012; Zeleny et al., 2014) have used this method to estimate shape memory effect in Heusler FSMAs, such as Ni$_2$MnGa and Mn$_2$NiGa.

In Figure 5, we can see that all X$_2$MnTi ($X = \text{Pd, Pt, Ag, Au, Cu, and Ni}$) compounds present a L$_2_1$-L$_1_0$ (possible martensitic transformation) under the effect of tetragonal distortion. In detail, we found that X$_2$MnTi (except for X = Au) compounds have two local minimum energies, one locates at $c/a < 1$ and the other locates at $c/a > 1$. The value of the local minimum energy with $c/a > 1$ is lower than that of the local minimum energy with $c/a < 1$, which means that the local minimum energy with $c/a > 1$ is the most stable state. Moreover, when $c/a < 1$, the local minimum energy of X$_2$MnTi (except for X = Au) is a metastable phase. For the Au$_2$MnTi compound, there is only one local minimum energy, which locates at $c/a > 1$. This local minimum energy is the most stable phase and there is no
FIGURE 5 | (A–F) Relationship between the \(c/a\) ratio and \(\Delta E (\Delta E = E_T - E_C)\) for \(X_2\)MnTi \((X = \text{Pd, Pt, Ag, Au, Cu, and Ni})\) compounds.

The calculated density of states of all-\(d\)-metal Heusler-type compounds \(X_2\)MnTi

We calculated their total density of states (TDOSs) and partial state density of states (PDOSs) \((\text{Han et al., 2019b})\) in cubic and tetragonal states, respectively, and we plotted them in Figures 7, 8.

We found that the TDOS of tetragonal structure around the Fermi level is softer than the cubic TDOS. For example, we can find some strong peaks (see the red lines) near the Fermi level in the spin-down direction and some small peaks in the spin-up direction (see the black line) for the cubic phase (see Figure 7A). Note that one of the contributions to the total energy is the band energy \(E_{\text{band}} = \int_{E_{\text{min}}}^{E_F} d\text{DOS}(E)E\); a reduction of the DOS near \(E_F\) in a tetragonal phase, in conjunction with the conservation of the integral for the number of valence electrons \(N_V = \int_{E_{\text{min}}}^{E_F} d\text{DOS}(E)\), often leads to a lower band energy and, thus, to a lower total energy for the tetragonal phase than for the cubic phase. As shown in Figure 7, we can find that, under the effect of the tetragonal strain, the energy states near \(E_F\) tend to decrease or move to a high energy level. We give two more detailed explanations as follows: (1) as shown in Figures 7A, C–F, we can clearly observe that the local energy states (blue lines) near the Fermi level of the tetragonal \(L_{10}\) phase are significantly lower than those of the cubic \(L_2\) phase (black lines). Thanks to the TDOSs above, the total energy of the system will be released, resulting in cubic–tetragonal transformation; (2) as shown in Figures 7A, C, we find that the peaks (red line) of the cubic structure near the Fermi level disappeared in the tetragonal structure; however, a very small valley at the same energy (see the pink line) occurred. The peak to valley transition of TDOS near the Fermi level also proves that the tetragonal \(L_{10}\) structure is more stable for \(X_2\)MnTi compounds.

The calculated density of states of all-\(d\)-metal Heusler-type compounds \(X_2\)MnTi

We calculated their total density of states (TDOSs) and partial state density of states (PDOSs) \((\text{Han et al., 2019b})\) in cubic and tetragonal states, respectively, and we plotted them in Figures 7, 8.

We found that the TDOS of tetragonal structure around the Fermi level is softer than the cubic TDOS. For example, we can find some strong peaks (see the red lines) near the Fermi level in the spin-down direction and some small peaks in the spin-up direction (see the black line) for the cubic phase (see Figure 7A). Note that one of the contributions to the total energy is the band energy \(E_{\text{band}} = \int_{E_{\text{min}}}^{E_F} d\text{DOS}(E)E\); a reduction of the DOS near \(E_F\) in a tetragonal phase, in conjunction with the conservation of the integral for the number of valence electrons \(N_V = \int_{E_{\text{min}}}^{E_F} d\text{DOS}(E)\), often leads to a lower band energy and, thus, to a lower total energy for the tetragonal phase than for the cubic phase. As shown in Figure 7, we can find that, under the effect of the tetragonal strain, the energy states near \(E_F\) tend to decrease or move to a high energy level. We give two more detailed explanations as follows: (1) as shown in Figures 7A, C–F, we can clearly observe that the local energy states (blue lines) near the Fermi level of the tetragonal \(L_{10}\) phase are significantly lower than those of the cubic \(L_2\) phase (black lines). Thanks to the TDOSs above, the total energy of the system will be released, resulting in cubic–tetragonal transformation; (2) as shown in Figures 7A, C, we find that the peaks (red line) of the cubic structure near the Fermi level disappeared in the tetragonal structure; however, a very small valley at the same energy (see the pink line) occurred. The peak to valley transition of TDOS near the Fermi level also proves that the tetragonal \(L_{10}\) structure is more stable for \(X_2\)MnTi compounds.
In Figure 8, we also show the PDOSs of each atom in the cubic and tetragonal phases. As shown in Figure 8A, the PDOS of the L2₁-type Ag orbitals is almost located in the energy areas of −4.5 to −6 eV. In this region, the spin-up and spin-down PDOSs of Ag-d are almost symmetrical, reflecting that the contribution of Ag atoms to the total magnetism is relatively small. In the energy range from −1.5 to −4 eV, two large energy peaks, which come from the Mn-d orbitals, can be found in the spin-up channel; however, the DOS in the spin-down is nearly flat. Above the Fermi level, the TDOS in the spin-up channel is coming from the Ti-d orbital, and the TDOS in the spin-down channel is arising from the hybridization between the Ti-d and Mn-d orbitals. A
similar phenomenon can also be found in L2₁-type Au₂MnTi as shown in Figure 8B.

As shown in Figures 8C–F, some large peaks can be obviously found below the Fermi level in the spin-up channel for the L₂₁-type X₂MnTi (X = Ni, Cu, Pd, Pt) alloys. The formation of these peaks owes to the hybridization between X-d and Mn-d orbitals. For Ni₂MnTi, the energy peak of the Ni-d orbital is larger than that of the Mn-d orbital around −1.2 eV; however, for the other three cases, the energy peaks of the X-d orbital are smaller than those of the X-d orbitals. Above the Fermi level, the TDOS is mainly coming from the Mn-d orbital in the spin-up channel and from the Mn-d and X-d orbitals in the spin-down channel.

As shown in Figure 8, near the Fermi level, PDOSs of the Mn atom will produce a strong spin splitting in two spin channels, and then result in strong magnetism. Thus, the total magnetism of X₂MnTi is mainly coming from the Mn atoms. We also exhibit the total and atomic magnetic moments in Tables 1, 2.

Finally, we calculated the phonon spectrum of the X₂MnTi (X = Pd, Pt, Ag, Au, Cu, and Ni) tetragonal L₁₀ phase by means of the force-constants method using Nanodcal code and the results are collected in Figure 9. In Figure 9, we can clearly see that there are no virtual frequencies in the phonon spectrum of the X₂MnTi, and the absence of virtual frequencies further confirms that their tetragonal L₁₀ states are theoretically stable. Unfortunately, the possible L₂₁–L₁₀ phase transition of X₂MnTi (X = Pd, Pt, Ag, Au, Cu, and Ni) has not been studied experimentally, and therefore, a comparison between the theoretical and experimental results cannot be shown in the current work. However, this investigation can help in understanding the physics in all-d-metal alloys. Moreover, we
FIGURE 8 | (A–F) PDOS diagrams of $X_2$MnTi ($X =$ Pd, Pt, Ag, Cu, Mn, and Ni) compounds in cubic and tetragonal phases.

TABLE 1 | Total and atomic magnetic moments for cubic $L_2^1$ type $X_2$MnTi ($X =$ Pd, Pt, Cu, Ni, Ag, Au).

| Compounds | $X_2$MnTi | $M_{Mn}$ ($\mu_B$) | $M_{Mn}$ ($\mu_B$) | $M_{X-1}$ ($\mu_B$) | $M_{X-2}$ ($\mu_B$) |
|------------|------------|---------------------|---------------------|---------------------|---------------------|
| Pd$_2$MnTi | 3.9295     | 3.792               | −0.108              | 0.123               | 0.123               |
| Pt$_2$MnTi | 4.17525    | 3.774               | 0.008               | 0.157               | 0.157               |
| Cu$_2$MnTi | 3.4385     | 3.415               | −0.117              | 0.069               | 0.069               |
| Ni$_2$MnTi | 3.90075    | 3.372               | −0.098              | 0.313               | 0.313               |
| Ag$_2$MnTi | 3.45875    | 3.79                | −0.376              | 0.022               | 0.022               |
| Au$_2$MnTi | 3.60475    | 3.76                | −0.215              | 0.03                | 0.03                |

TABLE 2 | Total and atomic magnetic moments, and $c/a$ ratio for tetragonal $L_10$ type $X_2$MnTi ($X =$ Pd, Pt, Cu, Ni, Ag, Au).

| Compounds | $X_2$MnTi | $M_{Mn}$ ($\mu_B$) | $M_{Mn}$ ($\mu_B$) | $M_{X-1}$ ($\mu_B$) | $M_{X-2}$ ($\mu_B$) | $c/a$ |
|------------|------------|---------------------|---------------------|---------------------|---------------------|-------|
| Pd$_2$MnTi | 3.18775    | 3.446               | −0.526              | 0.134               | 0.134               | 1.36  |
| Pt$_2$MnTi | 3.3185     | 3.429               | −0.396              | 0.143               | 0.143               | 1.35  |
| Cu$_2$MnTi | 1.77925    | 2.487               | −0.722              | 0.007               | 0.007               | 1.56  |
| Ni$_2$MnTi | 3.092      | 2.799               | −0.415              | 0.323               | 0.323               | 1.4   |
| Ag$_2$MnTi | 2.01       | 3.198               | −1.147              | −0.02               | −0.02               | 1.55  |
| Au$_2$MnTi | 2.42525    | 3.352               | −0.919              | −0.004              | −0.004              | 1.39  |
would like to point out that Bain paths are a sophisticated way to investigate the reversible transformation between the L2₁ and L1₀ phases during the tetragonal distortion. This method has been widely used to design new MSMAs, and some designed MSMAs have been experimentally verified, such as the Mn–Ni–Co–Ti system.

**CONCLUSIONS**

In this study, we investigated the phase transition and electronic structures of X₂MnTi (X = Pd, Pt, Ag, Au, Cu, and Ni) compounds based on first-principle calculations. First, we examined the L2₁ and XA competition of Heusler compounds.
X$_2$MnTi (X = Pd, Pt, Ag, Au, Cu, and Ni). The results show that the L2$_1$ type is the most stable ordered structure for these newly designed materials. Based on the L2$_1$ structure, we also compared the total energies of the L2$_1$, X$_2$MnTi system with different magnetic states, i.e., FM and AFM. We found that these compounds have lower ground state energy when in the FM state, that is, the most stable state of X$_2$MnTi is the FM state under the L2$_1$ structure. Subsequently, we studied the possible tetragonal transformation of Heusler compounds X$_2$MnTi (X = Pd, Pt, Ag, Au, Cu, and Ni) and found that these materials all feature a stable L1$_0$ tetragonal phase. The energy difference $\Delta E_{21}$ ($E_2 - E_1$) in X$_2$MnTi can be adjusted by a uniform strain. By analyzing the DOS diagram, it can be found that the magnetic moment of X$_2$MnTi mainly comes from Mn atoms, which is due to their strong spin splitting around $E_F$. The lower TDOS of the tetragonal L1$_0$ state near the Fermi level can be used to explain the stability of the tetragonal L1$_0$ state of X$_2$MnTi. Finally, we gave the phonon spectra of the tetragonal L1$_0$ phase of X$_2$MnTi (X = Pd, Pt, Ag, Au, Cu, and Ni) to further prove their stability.

**REFERENCES**

Ahmadian, F., and Salary, A. (2014). Half-metallicity in the Inverse Heusler compounds Sc$_2$MnZ (Z = C, Si, Ge, and Sn). *Intermetallics*. 46, 243–249. doi: 10.1016/j.intermetal.2013.11.021

Aksoy, S., Acet, M., Deen, P. P., Manosa, L., and Planes, A. (2009). Magnetic correlations in martenstic Ni-Mn based Heusler shape-memory compounds: neutron polarization analysis. *Phys. Rev. B*. 79:212401. doi: 10.1103/PhysRevB.79.212401

Alippi, P., Marcus, P. M., and Scheiffer, M. (1997). Strained tetragonal states and Bain paths in metals. *Phys. Rev. Lett*. 78:3892. doi: 10.1103/PhysRevLett.78.3892

Barkan, S. K., Banik, S., Shukla, A. K., Kamal, C., and Chakrabarti, A. (2007). Martenstic transition, ferrimagnetism and Fermi surface nesting in Mn$_2$NiGa. *EPL*. 80:57002. doi: 10.1209/0295-5075/80/57002

Birkel, C. S., Douglas, J. E., Lettiere, B. R., Seward, G., Verma, N., Zhang, Y., et al. (2013). Improving the thermoelectric properties of half-Heusler TiNiS$_n$ through inclusion of a second full-Heusler phase: microwave preparation and spark plasma sintering of TiNiS$_n$. *Phys. Chem. Chem. Phys*. 15:6990. doi: 10.1039/c3cp50918d

Cui, Z., Wu, B., Xin, R., Zhou, Q., and Feng, Y. (2019). Enhancing the half-metallicity of equiatomic quaternary Heusler compound CoFeCeGe via atomic doping. *Results Phys*. 15:102533. doi: 10.1016/j.rinp.2019.102533

Downie, R. A., Maclaren, D. A., Smith, R. I., and Bos, J. W. G. (2013). Enhanced thermoelectric performance in TiNiS$_n$-based half-Heuslers. *Chem. Commun*. 49:4184. doi: 10.1039/c3cc37121a

Gao, G. Y., and Yao, K. L. (2013). Antiferromagnetic half-metals, gapless half-metals, and spin gapless semiconductors: The DO$_2$$_1$-type Heusler compounds. *Appl. Phys. Lett.* 103:232409. doi: 10.1063/1.4840318

Ghannain, R., Ecker, V., Scholz, M., Gellesch, S., and Wurmelh, D. (2018). Carbon nanotube-assisted synthesis of ferromagnetic Heusler nanoparticles of Fe$_3$Ga (Nano-Gafilenl). *J. Mater. Chem. C*. 6:1255. doi: 10.1039/C7TC04618A

Graf, T., Felsen, C., and Parkin, S. S. P. (2011). Simple rules for the understanding of Heusler compounds. *Prog. Solid State Chem*. 39, 1–50. doi: 10.1016/j.progsolidchem.2011.02.001

Gschneidner Jr, K. A., Pecharsky, V. K., and Tsokol, A. O. (2005). Recent developments in magnetocaloric materials. *Rep. Progr. Phys*. 68:1479. doi: 10.1088/0034-4885/68/6/R04

Hafner, J. (2007). Materials simulations using VASP—a quantum perspective to materials science. *Comput. Phys. Commun*. 177, 6–13. doi: 10.1016/j.cpc.2007.02.045

Han, Y., Chen, Z., Kuang, M., Liu, Z., Wang, X., and Wang, X. (2019a). 171 Scandium-based full Heusler compounds: A comprehensive study of competition between XA and L2$_1$ atomic ordering. *Results Phys*. 12, 435–446. doi: 10.1016/j.rinp.2018.11.079

Han, Y., Wu, M., Feng, Y., Cheng, Z. X., and Wang, X. T. (2019b). Competition between cubic and tetragonal phases in all-d-metal Heusler compounds, X$_{2-n}$Mn$_n$V (X = Pd, Pt, Ag, Au, Cu, and Ni) and found that these materials all feature a stable L1$_0$ tetragonal phase. The energy difference $\Delta E_{21}$ ($E_2 - E_1$) in X$_2$MnTi can be adjusted by a uniform strain. By analyzing the DOS diagram, it can be found that the magnetic moment of X$_2$MnTi mainly comes from Mn atoms, which is due to their strong spin splitting around $E_F$. The lower TDOS of the tetragonal L1$_0$ state near the Fermi level can be used to explain the stability of the tetragonal L1$_0$ state of X$_2$MnTi. Finally, we gave the phonon spectra of the tetragonal L1$_0$ phase of X$_2$MnTi (X = Pd, Pt, Ag, Au, Cu, and Ni) to further prove their stability.

**DATA AVAILABILITY STATEMENT**

The raw data supporting the conclusions of this article will be made available by the authors, without undue reservation.

**AUTHOR CONTRIBUTIONS**

MW: conceptualization, methodology, software, and writing—original draft preparation. FZ and RK: software and writing—original draft preparation. MK and XW: supervision. All authors contributed to the article and approved the submitted version.

**FUNDING**

This study has been funded by the National Natural Science Foundation of China (Grant No. 11704315) and the Fundamental Research Funds for the Central Universities (Grant No. XDJK2019C112).

X$_2$MnTi (X = Pd, Pt, Ag, Au, Cu, and Ni). The results show that the L2$_1$ type is the most stable ordered structure for these newly designed materials. Based on the L2$_1$ structure, we also compared the total energies of the L2$_1$, X$_2$MnTi system with different magnetic states, i.e., FM and AFM. We found that these compounds have lower ground state energy when in the FM state, that is, the most stable state of X$_2$MnTi is the FM state under the L2$_1$ structure. Subsequently, we studied the possible tetragonal transformation of Heusler compounds X$_2$MnTi (X = Pd, Pt, Ag, Au, Cu, and Ni) and found that these materials all feature a stable L1$_0$ tetragonal phase. The energy difference $\Delta E_{21}$ ($E_2 - E_1$) in X$_2$MnTi can be adjusted by a uniform strain. By analyzing the DOS diagram, it can be found that the magnetic moment of X$_2$MnTi mainly comes from Mn atoms, which is due to their strong spin splitting around $E_F$. The lower TDOS of the tetragonal L1$_0$ state near the Fermi level can be used to explain the stability of the tetragonal L1$_0$ state of X$_2$MnTi. Finally, we gave the phonon spectra of the tetragonal L1$_0$ phase of X$_2$MnTi (X = Pd, Pt, Ag, Au, Cu, and Ni) to further prove their stability.
Muldawer, L. (1966). X-ray Study of Ternary Ordering of the Noble Metals in AgAuZn and CuAuZn. J. Appl. Phys. 37, 2062–2066. doi: 10.1063/1.1708670

O’Handley, R. C. (1998). Model for strain and magnetization in magnetic shape-memory compounds. J. Appl. Phys. 83, 3263–3270. doi: 10.1063/1.367094

Oikawa, K., Ota, T., Ohmori, T., Tanaka, Y., Morito, H., Fujita, A., et al. (2002). Magnetic and martensitic phase transitions in ferromagnetic Ni–Ga–Fe shape memory compounds. Appl. Phys. Lett. 81, 5201–5203. doi: 10.1063/1.1532105

Oikawa, K., Wulff, L., Iijima, T., Gejima, F., Ohmori, T., Fujita, A., et al. (2001). Promising ferromagnetic Ni–Co–Al shape memory compound system. Appl. Phys. Lett. 79, 3290–3292. doi: 10.1063/1.1418259

Özdemir Kart, S., Uluçanoglu, M., Karaman, I., and Cagin, T. (2008). DFT studies on structure, mechanics and phase behavior of magnetic shape memory alloys: Ni3MnGa. Phys. Status Solidi (a) 205, 1026–1035. doi: 10.1002/pssa.200776453

Perdew, J. P., Burke, K., and Ernzerhof, M. (1996). Generalized gradient approximation made simple. Phys. Rev. Lett. 77:3865. doi: 10.1103/PhysRevLett.77.3865

Perdew, J. P., Burke, K., and Ernzerhof, M. (1998). Perdew, burke, and ernzerhof reply. Phys. Rev. Lett. 80:891. doi: 10.1103/PhysRevLett.80.891

Populoh, S., Aguirre, M. H., Brunko, O. C., Galazka, K., Lu, Y., and Weidenkaff, A. (2012). High figure of merit in (Ti, Zr, Hf)NiSn half-Heusler compounds. Appl. Phys. Lett. 100:232406. doi: 10.1063/1.4730590

Qawasmeh, Y., and Bothina, H. (2012). Investigation of the structural, electronic, and magnetic properties of Ni-based Heusler alloys from first principles. J. Phys. Chem. B 116:12628. doi: 10.1021/jp209822z

Shigeta, I., Kubota, T., Sakuraba, Y., Kimura, S., and Awaji, S. (2018). Transport properties of epitaxial films for superconductor NbN and half-metallic Heusler compound Co2MnSi under high magnetic fields. Phys B Cond. Matt. 536, 310–313. doi: 10.1016/j.physb.2017.09.074

Silpawilawan, W., Kurosaki, K., Ohishi, Y., Muta, H., and Yamanaka. S. (2017). FeNiSb p-type half-heusler compound: beneficial thermomechanical properties and high-temperature stability for thermoelectrics. J. Mater. Chem. C 5:6677. doi: 10.1039/C7TC01570D

Singh, S., and Gupta, D. C. (2019). Lanthanum based quaternary Heusler compounds LaCo4C3X (X = Al, Ga): Hunt for half-metallicity and high thermoelectric efficiency. Results Phys. 13:102300. doi: 10.1016/j.rinp.2019.102300.

Skafkouros, S., Ozdogan, K., Sasioglu, E., and Galanakis, I. (2013). Search for spin gapless semiconductors: The case of inverse Heusler compounds. Appl. Phys. Lett. 102:22402. doi: 10.1063/1.4775599

Suzuki, R. O., and Kyono, T. (2004). Thermoelectric properties of Fe2TiAl Heusler compounds. J. Comp. Comp. 377, 38–42. doi: 10.1016/j.jallcom.2004.01.035

Tan, J. G., Liu, Z. H., Zhang, Y. J., Li, G. T., Zhang, H. G., Liu, G. D., et al. (2019). Site preference and tetragonal distortion of Heusler compound Mn-Ni-V. Results Phys. 12, 1182–1189. doi: 10.1016/j.rinp.2018.12.096

Ullakko, K., Huang, J. K., Kantner, C., Ohandley, R. C., and Kokorin, V. V. (1996). Large magnetic-field-induced strains in Ni3MnGa single crystals. Appl. Phys. Lett. 69, 1966–1968. doi: 10.1063/1.117637

Wang, X., Cheng, Z., Liu, G., Dai, X., and Bouhemadou, A. (2017). Rare earth-based quaternary Heusler compounds MCoVZ (M = Lu, Y; Z = Si, Ge) with tunable band characteristics for potential spintronic applications. IUCrJ. 4, 758–768. doi: 10.1107/S2052252517013264

Wang, X., Cheng, Z., Wang, J., and Liu, G. (2016). Recent advances in the Heusler based spin-gapless semiconductors. J. Mater. Chem. C. 4, 7176–7192. doi: 10.1039/C6TC01343K

Wang, X. T., Cheng, Z. X., Yuan, H. K., and Khenata, R. (2017). L21 and L22 ordering competition in titanium-based full-heusler alloys. J. Mater. Chem. C. 5:11559. doi: 10.1039/C7TC03999C

Wei, Z. Y., Liu, E. K., Chen, J. H., Li, Y., Liu, G. D., Luo, H. Z., et al. (2015). Realization of multifunctional shape-memory ferromagnets in all-d-metal Heusler phases. Appl. Phys. Lett. 107:222406. doi: 10.1063/1.4927058

Wurmehl, S., Fecher, G. H., Kandpal, H. C., Ksenofontov, V., Felser, C., Lin, H. J., et al. (2005). Geometric, electronic, and magnetic structure of Co2FeSi: Curie temperature and magnetic moment measurements and calculations. Phys. Rev. B. 72:184434. doi: 10.1103/PhysRevB.72.184434

Xue, Q. Y., Liu, H. J., Fan, D. D., Cheng, L., Zhao, B. Y., and Shi, J. (2016). LaPtSb: a half-Heusler compound with high thermoelectric performance. Phys. Chem. Chem. Phys. 18:17912. doi: 10.1039/C6CP03112G

Zeleny, M., Sozinov, A., Straka, L., Björkman, T., and Nieminen, R. M. (2014). First-principles study of Co-and Cu-doped Ni3MnGa along the tetragonal deformation path. Phys. Rev. B. 89:184103. doi: 10.1103/PhysRevB.89.184103

Conflict of Interest: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

Copyright © 2020 Wu, Zhou, Khenata, Kuang and Wang. This is an open-access article distributed under the terms of the Creative Commons Attribution License (CC BY). The use, distribution or reproduction in other forums is permitted, provided the original author(s) and the copyright owner(s) are credited and that the original publication in this journal is cited, in accordance with accepted academic practice. No use, distribution or reproduction is permitted which does not comply with these terms.