A Study on the Transition of Copper Oxide by the Incorporation of Nitrogen

Song-Yi Ahn, Kyung Park, Daehwan Choi, Jozeph Park*, Yong Joo Kim and Hyun-Suk Kim

1 Department of Materials Science and Engineering, Chungnam National University, Daejeon 34134, Korea; dksthddl123@naver.com
2 Semiconductor Process Laboratory, WONIK IPS, Gyeonggi-do 17709, Korea; kyungpark@yonsei.ac.kr
3 School of Integrated Technology, Yonsei University, Incheon 21983, Korea; maximum@yonsei.ac.kr
4 Department of Material Science and Engineering, Korea Advanced Institute of Science and Technology, Daejeon 34141, Korea
5 Biosystems Machinery Engineering, Chungnam National University, Daejeon 34134, Korea
* Correspondence: jozeph.park@gmail.com (J.P.); babina@cnu.ac.kr (Y.J.K.); khs3297@cnu.ac.kr (H.-S.K.)

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Abstract: In the present study, the effects of nitrogen incorporation on the transition of a p-type copper oxide semiconductor are investigated. The properties of sputtered copper oxide and nitrogen-incorporated copper oxide are evaluated and compared at various nitrogen gas flow rates. The results indicate that the addition of nitrogen results in an increased optical bandgap, accompanied by significantly reduced tail states compared to pristine copper oxide. In addition, X-ray diffraction and X-ray photoelectron spectroscopy reveal that the incorporation of nitrogen stimulates the transition from copper (II) oxide to copper (I) oxide.

Keywords: nitrogen incorporation; CuON; p-type oxide semiconductor; reactive sputtering; thin-film transistors

1. Introduction

Recently, thin-film transistors (TFTs) based on metal oxide semiconductors such as In–Ga–Zn–O (IGZO) have attracted considerable attention for pixel-switching devices in flat-panel displays owing to their relatively high performance and stability compared to amorphous silicon (a-Si) [1–3]. Commercial products incorporating TFT backplanes based on n-type IGZO are already available in the market. On the other hand, relatively few reports are available on p-type oxide semiconductors.

Candidate p-type oxide semiconductors include copper oxide (CuOₓ), tin oxide (SnOₓ), and nickel oxide (NiOₓ) [3–7]. Among them, copper oxide is a non-toxic, cost-effective material that may be deposited by various techniques such as sputtering, chemical vapor deposition, and even the sol–gel method [8–11]. As a transition metal, copper oxide may exhibit two different phases, namely cupric oxide (copper (II) oxide, CuO) and cuprous oxide (copper (I) oxide, Cu₂O). The monoclinic CuO and cubic Cu₂O have optical bandgap energy values around 1.9–2.1 eV and 2.1–2.6 eV, respectively [12,13]. Generally, most amorphous oxide semiconductors have significantly low hole mobility compared to electron mobility, because of the localized valence band (VB) states resulting from highly directional O 2p orbitals. However, copper oxide exhibits a quite different orbital configuration of the VB states. The Cu 3d and O 2p orbitals are located at nearly the same energy level near the valence band maximum (VBM), and are fully occupied by hybridization. Such a configuration forms less-localized VBM states, creating relatively fast pathways for hole carriers [14–16]. Charge transport in copper oxide occurs mainly via charged Cu vacancies, which generate acceptor levels near 0.3 eV above the VBM [17,18].
Electronics 2019, 8, 1099

obtain high-performance, copper oxide-based p-type thin-film transistors, the former studies generally reported on the effect of annealing temperature, oxygen flow ratio during growth, and external dopant concentration [19–22]. Especially, the incorporation of nitrogen was observed to improve the electrical properties of copper oxide, yet the exact mechanism was not elucidated.

In the present work, the properties of CuO and N-incorporated CuO (CuON) thin films and the related TFTs are investigated. The results indicate that CuON has significantly decreased tail states compared to pristine CuO, as well as increased optical bandgap energy. Moreover, X-ray diffraction (XRD) and X-ray photoelectron spectroscopy (XPS) examinations reveal that the nitrogen incorporation in copper oxide stimulates the transition from copper (II) oxide to copper (I) oxide.

2. Experimental Section

Cu₉O and CuON thin films were deposited by reactive radio frequency (RF) magnetron sputtering at room temperature with a 3-inch diameter metal Cu target (99.99%). The applied RF power and gas mixing ratio of Ar/O₂ were fixed at 80 W and 10:10 (sccm/sccm), respectively. The nitrogen gas flow rates were varied from 0 to 50 sccm. The working pressure values also varied from 7 to 22 mTorr as the amount of nitrogen gas was increased. The microstructures of Cu₉O and CuON thin films were analyzed by grazing incidence angle X-ray diffraction (GIAXRD) using a Cu kα radiation (D8 DISCOVER, Bruker AXS, Billerica, MA, USA). The surface morphologies were studied by atomic force microscopy (AFM). The chemical bonding states of nitrogen and oxygen components were examined by X-ray photoelectron spectroscopy (XPS) (K-ALPHA ESCA SYSTEM, ThermoFisher Scientificz, Waltham, MA, USA). The optical transmittance and bandgap values were extracted using a Transmittance UV-vis Spectrometer (S-3100, SCINCO, Seoul, Korea) in the range of 200 to 1000 nm. Hall measurements were also carried out to observe the electrical characteristics of thin films (HMS-5500, ECOPIA, Gyeonggi-do, Korea).

Cu₉O and CuON TFTs with bottom-gated structures were fabricated on highly doped p-type Si substrates with thermally grown 100-nm thick SiO₂ gate dielectrics. After depositing the Cu₉O or CuON active layers, post-annealing was conducted at 300 °C for 2 h in a vacuum ambient by a rapid thermal annealing system (RTA). Then, the source–drain electrodes were formed using a shadow mask, by depositing a 100-nm thick nickel (Ni) film by thermal evaporation. The width and length of the resulting devices were 800 µm and 200 µm, respectively. The transfer characteristics were analyzed using an HP 4156B semiconductor parameter analyzer (HEWLETT PACKARD, CA, USA) in the dark under ambient conditions.

3. Results and Discussions

The XRD patterns of pristine Cu₉O exhibit intense (−111) and (111) peaks of a monoclinic structure at 2θ = 36 and 38°, respectively in Figure 1a. After annealing, the same peaks persist, as shown in Figure 1b. Similarly, the as-deposited CuON film also exhibits identical peaks; however, the (−111) peak is more intense compared to those of the as-deposited and annealed Cu₉O films (Figure 1c). On the other hand, the annealed CuON film exhibits a quite different structure. The corresponding diffractogram in Figure 1d exhibits the (111) and (200) peaks of Cu₂O (N₂ 50 sccm). No apparent reaction between copper and nitrogen seems to have taken place, since even at relatively high nitrogen gas flow rates, no peaks related to compounds such as Cu₃N are detected. At this point, it is suggested that nitrogen inhibits the oxidation of CuO and promotes the formation of the Cu₂O phase. When nitrogen is incorporated in the plasma environment, nitrogen may play a role of removing oxygen gas as follows: 1/2O₂ + N₂ → N₂O or O₂ + N₂ → 2NO. Figure 2 shows the AFM surface morphologies of Cu₉O and CuON layers. In the Cu₉O film, the root mean square (RMS) surface roughness is 0.83 nm, while CuON films grown with different nitrogen gas flow rates all exhibit relatively smooth surfaces with RMS roughness values no greater than 0.57 nm.
The optical transmittance curves of Cu$_x$O and CuON layers were collected over a wavelength range of 200–1000 nm, as shown in Figure 3a. Figure 3b shows how Tauc’s law is used to extract the optical bandgap values of 1.99 and 2.49 eV for Cu$_x$O and CuON, respectively. After annealing, the optical bandgap values increased slightly up to 2.00 and 2.59 eV for Cu$_x$O and CuON, respectively. Therefore, the incorporation of nitrogen appears to affect the optical bandgap. Figure 3c shows the absorption coefficient spectra as a function of photon energy in the Cu$_x$O and CuON films. The tail state extending in the bandgap is called the Urbach tail, and its slope reflects the number of localized trap states. The small slope of the Urbach tail suggests that it has a large amount of localized trap state...
in the optical bandgap [20]. As the absorption spectra of pristine and annealed films in the subgap region show, CuON (i.e., the region of the green circle marked with $E_g < 2.59$ eV) layers exhibit smaller subgap defect absorption than Cu$_x$O (i.e., the region of the blue circle marked with $E_g < 2.00$ eV) films. At this point, it is highly likely that the presence of nitrogen results in decreased tail states.

Figure 3. Cont.
XPS analyses were performed to observe the chemical bond properties in the films. The peak positions were calibrated using the C 1s peak, of which the standard binding energy is about 284.5 eV. Figure 4 shows the Cu 2p\textsubscript{1/2} spectra of annealed Cu\textsubscript{x}O and CuON. Both as-deposited Cu\textsubscript{x}O and CuON thin films exhibit shoulder peaks around 954 eV and satellite peaks near 962 eV. Such satellite and shoulder peaks originate from copper (II) oxide [22–24]. The intensities of the satellite and shoulder peaks in CuON decrease largely after annealing, which strongly suggests that some transition from monoclinic CuO to cubic Cu\textsubscript{2}O has occurred. Such results are also in agreement with the XRD results shown in Figure 1. Figure 5a–d shows the O 1s peak spectra of the Cu\textsubscript{x}O and CuON films. The peaks were resolved into two sub-peaks from metal oxygen bonds (M–O) and oxygen-related defects bond such as oxygen deficient sites (V\textsubscript{O}) and contamination species (OH\textsuperscript{-}). Generally, it is well-known that the M–O, oxygen-related defects peaks in metal oxides are located near 530.6 eV and 531.9 eV, respectively [19,25,26]. In Cu\textsubscript{x}O films, the O 1s sub-peak spectra did not change much after annealing, which suggests that no transition has taken place. However, the addition of nitrogen results in an increased oxygen-related defects peak portion from 20.9% to 34.9%, as is observed in pristine Cu\textsubscript{x}O and CuON, respectively. It suggests that nitrogen incorporation may disturb Cu–O bonds and increase disordering or O-related defect density. However, as the CuON layers are annealed, the oxygen-related defects peak portion decreases considerably from 34.9% to 17.7%. While the migration of nitrogen forming Cu–N bonds may have contributed to the decrease in the intensity of the oxygen-related defects, the transition from CuO to Cu\textsubscript{2}O is also a probable factor. Figure 5e shows the N 1s peak of the pristine and annealed CuON thin films. The peak at 398.08 eV indicates the presence of Cu–N bonds [20]. In contrast, Cu–N bonds are not detected in pristine CuON, which implies that the incorporated nitrogen atoms are most likely to exist in the form of weakly bonded interstitials.
Figure 4. X-ray photoelectron spectroscopy (XPS) Cu 2p$_{1/2}$ peaks of (a) as-deposited Cu$_x$O, (b) annealed Cu$_x$O, (c) as-deposited CuON, and (d) annealed CuON.

The relative atomic compositions of Cu$_x$O and CuON films are indicated in Table 1. Only the annealed CuON film exhibits a Cu:O composition ratio of 2:1, while the other samples exhibit Cu:O ratios near 1:1. In accordance with the XRD results, it may be concluded that the incorporated nitrogen promotes the transition from CuO to Cu$_2$O after thermal treatment.

Table 1. The atomic compositions of Cu$_x$O and CuON films obtained by XPS analysis.

| Experimental Condition | Cu   | O     | N    |
|------------------------|------|-------|------|
| N$_2$ 0: as dep        | 48.98| 51.01 | -    |
| N$_2$ 0: 300 °C 2 h    | 57.66| 41.88 | 0.44 |
| N$_2$ 50: as dep       | 50.52| 49.47 | -    |
| N$_2$ 50: 300 °C 2 h   | 69.77| 28.74 | 1.48 |
Figure 4. X-ray photoelectron spectroscopy (XPS) Cu 2p 1/2 peaks of (a) as-deposited Cu xO, (b) annealed Cu xO, (c) as-deposited Cu ON, and (d) annealed Cu ON. 

(a) N₂ 0 sccm: as-deposited 

- Raw 
- Fit 
- O1: M-O Bond 
- O2: O-related defects Bond 

Normalized Intensity (a.u.)

Binding Energy (eV)

79.1% 20.9%

(b) N₂ 0 sccm: 300°C 2h 

- Raw 
- Fit 
- O1: M-O Bond 
- O2: O-related defects Bond 

Normalized Intensity (a.u.)

Binding Energy (eV)

78.5% 21.5%

(c) N₂ 50 sccm: as-deposited 

- Raw 
- Fit 
- O1: M-O Bond 
- O2: O-related defects Bond 

Normalized Intensity (a.u.)

Binding Energy (eV)

65.1% 34.9%

Figure 5. Cont.
Figure 5. XPS O 1s peaks of (a) as-deposited Cu$_x$O, (b) annealed Cu$_x$O, (c) as-deposited CuON, and (d) annealed CuON. (e) XPS N 1s spectra of as-deposited CuON and annealed CuON.

Table 2 shows the Hall measurement results of Cu$_x$O and CuON thin films, which exhibited p-type behavior. As the nitrogen flow rate is increased during CuON growth, the carrier concentration values decrease from $2.24 \times 10^{20}$ to $3.59 \times 10^{14}$/cm$^3$, whereas the Hall mobility increases from 0.33 to 49.8 cm$^2$/Vs. Generally, Cu$_2$O exhibits relatively higher hole transport properties than CuO, even at relatively low carrier concentrations. Here, the increasing Hall mobility may be attributed to the presence of nitrogen, which helps to maintain the Cu$_2$O phase.

Table 2. Hall measurement results of CuON films grown at various N$_2$ flow rates.

| N$_2$ (sccm) | Carrier Concentration (cm$^{-3}$) | Hall Mobility (cm$^2$/V·s) |
|-------------|---------------------------------|-----------------------------|
| 0           | $2.24 \times 10^{20}$           | 0.33                        |
| 10          | $9.86 \times 10^{16}$           | 4.09                        |
| 20          | $1.49 \times 10^{15}$           | 32.2                        |
| 50          | $3.59 \times 10^{14}$           | 49.8                        |
Figure 6a shows the transfer curves of CuxO and CuON TFTs with an optimized channel layer thickness at 10 nm. The gate-to-source voltage (VGS) is measured in the range of −30 to 30 V, and the drain-to-source voltage (VDS) is fixed at −30 V. The conductive characteristics of CuxO TFTs may be explained by the relatively high carrier concentration. The on–off current ratio and the field effect mobility of CuxO TFTs are 1.5 and 2.7 × 10−4 cm2/Vs, respectively. In contrast, the CuON TFTs show improved switching characteristics, with an on–off current ratio and field effect mobility of 7.4 and 6.93 × 10−3 cm2/Vs, respectively. It is highly anticipated that N 2p orbitals may further enhance the delocalization and hybridization of valence band states, which would enhance the p-type transport properties. CuxO and CuON TFTs have poor electrical properties due to the charge trap density between the active layer and the dielectric [19]. The output curve characteristics of the CuxO and CuON TFTs are shown in Figure 6b. While only linear properties are observed in the CuxO TFT, the CuON device exhibits clear pinch-off and current saturation without current crowding.

![Graph of transfer curves](image)

**Figure 6.** (a) Characteristics of CuxO and thin-film transistors (TFTs) transfer curve according to N2 flow rates. (b) Output curves of the CuxO and CuON TFTs.

### 4. Conclusions

In the present work, the properties of CuO and CuON thin films and the related TFTs were investigated. Structural and chemical analyses indicate that the incorporation of nitrogen promotes the formation of the CuxO phase. In the transmittance spectra, relatively small band tail states are observed in CuON compared to CuxO, indicating that CuON is less defective. Also, CuON films exhibit higher hole mobility than pristine CuxO. Such electrical characteristics are also reflected in the transfer and output curves of the corresponding TFT devices. The above results indicate that the nitrogen incorporation is an effective method not only to obtain a stable CuxO structure, but also to enhance the p-type charge transport properties.

**Author Contributions:** S.-Y.A. performed the experimental work, analyzed data and wrote the paper; XPS measurement was performed by D.C.; K.P. assisted with XPS data analysis; J.P. and Y.J.K. assisted with writing and editing the paper; H.-S.K., supervised this work.

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