Ceriodaphnia dubia as a Potential Bio-Indicator for Assessing Acute Aluminum Oxide Nanoparticle Toxicity in Fresh Water Environment

Sunandan Pakrashi1, Swayamprava Dalai1, Ahmed Humayun1, Sujay Chakravarty2, Natarajan Chandrasekaran1, Amitava Mukherjee1*

1 Centre for Nanobiotechnology, VIT University, Vellore, India, 2UGC-DAE CSR, Kalpakkam Node, Kokilamedu, India

Abstract

Growing nanomaterials based consumer applications have raised concerns about their potential release into the aquatic ecosystems and the consequent toxicological impacts. So environmental monitoring of the nanomaterials in aqueous systems becomes imperative. The current study reveals the potential of Ceriodaphnia dubia (C. dubia) as a bio-indicator for aluminium oxide nanoparticles in a fresh water aquatic ecosystem where it occupies an important ecological niche as a primary consumer. This study aims to investigate the aluminium oxide nanoparticle induced acute toxicity on Ceriodaphnia dubia in a freshwater system. The bioavailability of the aluminium oxide nanoparticles has been studied with respect to their aggregation behavior in the system and correlated with the toxicity endpoints. The oxidative stress generated by the particles contributed greatly toward their toxicity. The crucial role of leached aluminium ion mediated toxicity in the later phases (48 h and 72 h) in conjunction with the effects from the nano-sized particles in the initial phases (24 h) puts forth the dynamics of nanotoxicity in the test system. The internalization of nanoparticles (both gross and systemic uptake) as substantiated through the transmission electron microscopy (TEM) and inductively coupled plasma optical emission spectral (ICP-OES) analysis was another major contributor toward acute toxicity. Concluding the present study, Ceriodaphnia dubia can be a promising candidate for bio-monitoring the aluminium oxide nanoparticles in a fresh water system.

Introduction

The present scale at which the nanotechnology industry is growing, it is estimated that it will be worth nearly a trillion dollar by the year 2015 [1], optimistic forecasts suggest the production of metal oxide nanoparticles to rise from 270, 041 tons in 2012 to 1663, 168 tons by 2020 [2]. Engineered nanoparticles find application in variety of products viz. sunscreen fillers [3], bio-imaging probes [4], photovoltaic cells [5], drug delivery [6], gene delivery [7]. Alumina, a relatively inert oxide of aluminium leads the nano-material manufacture by an estimated share of 20% of the total market production [8], it has potential uses in varied fields such as abrasives [9], flooring material [10], high performance paints [11], ultrafiltration membranes [12], jet fuel [13], rocket propulsion fuel [14], heat enhancing nano fluids [15]and vaccines [16].

With increasing proliferation in the usage it is certain that aluminium oxide nanoparticles will ultimately find a way into the biosphere [17], chemtrails left by jet fuel have already raised concerns over their role increased incidences of Alzheimer disease [18]. Several relevant studies concerning aluminium oxide nanoparticles toxicity include its detrimental effects on prokaryotic microorganisms [19], microalgae species [20], aquatic cladocerans [21], earthworms [22], nematodes [23], zebra fish [24] and cell lines [25]. Despite the growing market size of aluminum oxide and its proven toxicity record we feel that there is a dearth of available literature concerning its toxicity in the aquatic environment and a significant knowledge gap exists between time dependent monitoring of bioavailability and resulting toxicity of nanoparticles [26].

Among the fresh water aquatic invertebrates, Daphnia magna and Ceriodaphnia dubia are recognized as popular choices of test organisms for eco-toxicological studies [27], ecologically they serve as vital links between the producers (algae) and secondary consumers (fish and fish larvae). In several studies C. dubia has been suggested to be a significantly better ecotoxicity test organism due to higher sensitivity and shorter growth span [28], [29]. The detection and monitoring of aluminium oxide contamination can be a challenging task in complex environmental conditions. The sensitivity towards aluminium oxide and a strong dose and exposure dependence along with wide ecological distribution of daphnids in fresh water aquatic systems makes it a promising candidate for bio indicator [30].

Previously we have analyzed the toxic impact of aluminium oxide and titanium dioxide nanoparticles on the freshwater bacterial and algal isolates (Bacillus licheniformis, Chlorella and Scenedesmus sp) under environmentally relevant conditions [31]
Aqueous ecosystems for highly variable and complex nanoparticle.

C. dubia has been taken into account. Importantly, this study indicates that aluminum oxide nanoparticles in terms of the dynamic nature of solubilization of the aluminum ions from aluminum oxide, reactive oxygen species (ROS) and amount of internalized aluminum oxide were accounted for observed toxic response. Direct visualization of the damaged tissue linings along the digestive tract under transmission electron microscope validated the preliminary findings. Considering the ecological niche of C. dubia in fresh water aquatic environment, their toxic response can further be extrapolated to analyze the comprehensive nanotoxicity impact on the aquatic ecosystem making C. dubia as a potential bio indicator. Thus, it is a first of its kind of report, detailing toxic response of aluminum oxide nanoparticles in terms of the dynamic nature of nanotoxicity where their bioavailability and possible solubilization has been taken into account. Importantly, this study indicates that C. dubia can serve as a potential bio-monitoring tool for relevant aqueous ecosystems for highly variable and complex nanoparticle mediated toxicity.

Materials and Methods
Isolation, Characterization and Maintenance of Ceriodaphnia Dubia

The fresh water samples were collected from VIT Green Lake (12°58’10”N, 79°9’37”E) and analyzed for the presence of planktonic crustaceans which were isolated using 5 mL pasteur pipette (Sigma Aldrich, USA) and maintained under laboratory conditions in sterilized lake water medium prepared following the procedure as described in our earlier report [19]. Chlora sp at a concentration of 10⁶ cells/mL was added to supplement as the nutrition source in the medium. Subsequently the planktonic isolates were analyzed under light microscope at 10x and 40x magnifications for morphological identification and screening of the planktonic species as the organism of interest—Ceriodaphnia dubia, which was isolated into separate beakers containing similar lake water medium and was subcultured on a regular basis to ensure species uniformity of the subcultured colony. Ceriodaphnia dubia constituted ~40% of various Daphnia species found in the ‘VIT Green Lake’ water. Thus, it was considered to be a suitable representative candidate for the study. The requisite conditions of 20°C temperature and light: dark cycle of 14:10 were maintained in a temperature controlled chamber (I.L.E Co., India) and fluorescent light at 3000 Lux was used to maintain the light phase. The average life span of the daphnids was noted to be 20–28 days under the specified conditions.

No specific authorization was necessary for collecting samples for the study as the lake is situated inside the University campus. The experiments involved were carried out under the confines of the laboratory and due care was exercised not to contaminate or disrupt the lake water ecosystem.

Procurement of Nanoparticles

Aluminum oxide (Al2O3) was procured from Sigma Aldrich (St. Louis Missouri; CAS number 1344-28-1) in the form of dry powder. The suppliers product specifications are as follows - gamma phase alumina nanoparticles, particle size <50 nm, surface area 35–43 m²/g (BET). All other chemicals used in this study were of analytical grade. No specific information regarding surface coatings meant for stabilizing the particles were available from the manufacturer.

Characterization of as Received Nanoparticles

Crystalline configuration of the nanoparticles was typified by X-ray Diffraction analysis (D8 Advanced X-ray Diffractometer, Bruker, Germany). 250 mg of dry aluminum oxide nanoparticle was deposited in a sample holder and was scanned in the range of 10°–100° using 2.2 kW Cu anode radiation at a wavelength of 1.54 A° produced by a Ceramic X-ray tube. Scherrer formula was employed to compute the diffraction pattern and crystalline lattice geometry. The obtained data was matched with database of Joint Committee on Powder Diffraction Standards (JCPDS).

The size and shape of the procured aluminum oxide nanoparticles were analyzed and validated through transmission electron microscopy. A uniform dispersion of aluminum oxide nanoparticle of 10 µg/mL concentration was prepared in the Millipore filtered water and subjected to sonication using an ultrasonic processor with a maximum output of 130 W and frequency of 20 kHz (Sonics Corp., USA). A drop of the dispersion was taken on a lacey copper grid and dried in a dust free chamber for 4 h and was subjected to transmission electron microscopy (Philips CM12 Transmission Electron Microscope, Netherlands).

The particles were subjected to Fourier transform infrared (FT-IR) spectroscopic techniques to ascertain the surface functional groups. About 200 mg of aluminum oxide nanoparticles were mixed thoroughly with KBr and a pellet was made using a hydraulic press. The pellet was carefully transferred to the spectrophotometer (IRAffinity-1, Shimadzu, Japan) and analyzed for the presence of surface functional groups.

Stability Analysis of Nanoparticle Dispersion

Colloidal stability of nanoparticles in dispersion is of prime importance in terms of toxicity evaluation. A detailed stability analysis for the entire test concentrations ranging from 10–120 µg/mL were carried out at the stipulated time intervals of 0, 24 and 48 h. As the mean hydrodynamic diameter (MHD) was found to be above 1000 nm, the analysis was stopped at 48 h due to low resolving capacity of the particle size analyzer for particle size above 1000 nm.

The aliquots were taken from the dispersions of 20, 40, 60, 80, 100 and 120 µg/mL at 0, 24 and 48 h intervals. The measurements were conducted for 3 min using particle size analyzer (90 plus Particle Size Analyzer, Brookhaven instruments corporations, Brookhaven).
USA). BI-DLSW software was used to compute the mean hydrodynamic diameter (MHD) corresponding to the autocorrelation function of light scatter intensity of the solvated aluminum oxide nanoparticle surface. The concentration dependent aggregation profile involving dispersions of 20 μg/mL to 120 μg/mL was studied for 48 h. The analysis was stopped at this time point because it had reached beyond 1000 nm and particle size analyzer is not sensitive to higher size ranges. Furthermore, such dimensions it is expected to behave more like bulk aluminum oxide particles.

**Figure 2. Transmission Electron Microscopic analysis of aluminium oxide nanoparticles.** Transmission electron micrograph confirmed the spherical shape of the aluminium oxide nanoparticles with particle size ranging from 40 nm to 100 nm diameters. (n = 3).
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**Figure 3. Particle aggregation profile of aluminium oxide nanoparticles.** Time dependent variation in mean hydrodynamic diameter (MHD) of aluminium oxide nanoparticles was observed with respect to particle concentration. (n = 3).
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Bioavailability of Aluminum Oxide Nanoparticles during Exposure

Bioavailability is an important measure influencing the toxicity response of any substance. This becomes even more relevant in case of insoluble systems like aluminum oxide where the availability of the particle is dynamic in nature. This necessitates a real time analysis of available fraction of the added concentration.

About 5 mL of aliquots were collected at 0, 24, 48 and 72 h from the middle layer of the container which represents the median concentration of the system. This was digested using excess of 1 N nitric acid followed by subsequent dilutions and filtration through 0.45 μm membrane filter to ensure interference of any undigested particle. These samples were analyzed using Inductively Coupled Plasma-Optical Emission Spectrometry (ICP-OES, Perkin Elmer Optima 5300 DV, USA). This experiment was carried out in triplicates and repeated thrice to ensure reproducibility of the results obtained.

Toxicity Analysis

In accordance with the OECD test guidelines 202 [35], acute toxicity tests of prolonged test duration 72 h were carried out. Experimental setup consisted of 100 mL glass beakers each containing 5 healthy, lab bred C. dubia not more than 24 h old dispersed in about 20 mL of sterile lake water having aluminum oxide nanoparticles in increasing concentrations of 20, 40, 60, 80, 100 and 120 μg/mL respectively. All the experiments were conducted in triplicates and repeated more than three times to ensure reproducibility. Data have been presented as a relative survival considering untreated group as 100%.

C. dubia in batches of five organisms each was taken in each glass beaker containing the sterile lake water. At the end of 24, 48 and 72 h, the organisms were removed and the survivors were counted. Figure 5 shows the relative survival of C. dubia at different concentrations of aluminum oxide nanoparticles.

Bioavailability of Aluminium Oxide Nanoparticles

Figure 4. Bioavailability profile of aluminium oxide nanoparticles. The available concentration of aluminium oxide nanoparticles in the test matrix is seen to decrease with respect to time. The rate of reduction of bioavailability is faster as the concentration of the nanoparticle increases. (n = 3).

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Figure 5. Effect of aluminium oxide nanoparticles on C. dubia survival. The relative survival of C. dubia was shown to decrease with respect to time. With nanoparticle concentration, survival rate increased which suggests reduced toxicity of nanoparticles probably due to aggregation. (n = 3).

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72 h, immobile organisms were isolated under compound microscope. As per the Annex 201 of OECD 211 protocol [30] an inanimate test organism if unresponsive to physical stimulus for more 15 seconds was considered dead. Thus the toxicity endpoints expressed as LC50 were calculated. Throughout the toxicity analysis the mortality in untreated group was found to be less than 10%.

Oxidative Stress Analysis

Though Reactive Oxygen Species (ROS) produced as a byproduct of normal cell metabolism, in stressful conditions it may lead to excessive production leading to extensive cell membrane damage [36], [37]. Intracellular ROS concentration was determined using 2,7-dichlorofluorescin diacetate (DCFH-DA). It interacts with intracellular ROS leading to formation of a highly fluorescent complex called dichlorofluorescein, 5 μL DCFH-DA dye was added to the tissue homogenate of treated C. dubia and incubated at 25°C for 30 min under dark conditions. Resulting fluorescence was detected at 485 nm excitation and an emission at 532 nm using a fluorescence spectrophotometer (SL174, 191 ELICO).

Dissolution of Soluble Aluminum during Exposure

Dispersions of aluminium oxide nanoparticles (20, 40, 80, 100 and 120 μg/mL) in the filtered lake water were incubated in a shaker at room temperature. After 24, 48 and 72 h intervals of incubation, the dispersion was centrifuged at 12000 rpm at 4°C for 20 min, and then filtered through 0.1 μm and 10 kDa membrane filter with a pore size of ~2 nm [38], [19], [39]. As the particles present in the system were well above 100 nm, it is expected that the filtration will remove all particulate aluminum oxide from the suspension. However, a particle size analysis confirmed the complete removal of aluminium oxide nanoparticles from the system. The soluble aluminium content in the suspension was measured at a wavelength of 308.22 nm using ICP-OES (Perkin Elmer Optima 5300 DV, USA).

Gross Internalization of Aluminum Oxide Nanoparticles

Gross internalization of aluminum oxide nanoparticles into C. dubia was assessed after 72 h exposure. After interaction, the organisms were isolated and washed twice in Millipore filtered water to remove loosely adhering nanoparticles from the surface. The aluminium content in acid digested tissue samples was

Figure 6. Effect of labile aluminium leached from aluminium oxide nanoparticles. (A) Dissolution of ions from the aluminium oxide nanoparticles showed an increasing trend with time, but the release kinetics is not concentration dependent. (B) The relative survival of C. dubia upon exposure to labile aluminium was quantified. It showed no effect up to 24 h exposure, whereas, 48 and 72 h exposures showed a significant impact on the survival compared to the untreated group.

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analyzed using ICP-OES (PerkinElmer 270 Optima 5300 DV, USA). The equivalent amount of aluminum oxide was calculated from the obtained metallic aluminum content.

**Systemic Uptake of Aluminum Oxide Nanoparticles**

Internalization of the nanoparticles was further probed to provide an account for residual amount of aluminum oxide which remained unchanged even after 48 h depuration. After specified 72 h exposure, organisms were collected and transferred into sterile lake water and subjected to depuration step for 48 h duration. This provides sufficient time for clearance of non-internalized nanoparticles. Depurated samples were subjected to ICP-OES analysis after digesting the tissues. The equivalent concentration of aluminium oxide was computed from metallic aluminium content obtained through the analysis.

**Transmission Electron Microscopic Analysis**

The tissues of *Ceriodaphnia dubia* were noted to degrade rapidly following their death. Due to this reason, live organisms were collected for microscopic examination. Highest exposure concentration of 120 μg/mL and maximum duration of 72 h was selected for microscopy due to increased probability of observing altered or damaged structures indicating toxic impact. After 72 h exposure as many as 40 organisms were collected and fixed using glutaraldehyde followed by repeated washing. These samples were then stained using toluidine blue and semi thin (0.5 μm to 1 μm) sections were prepared. Further sectioning of the zones of interest was done and ultrathin sections of 3 nm to 10 nm thickness were observed using transmission electron microscope (Philips, Netherlands).

**Statistics Applied**

To ensure statistical validity of the findings, experiments were carried out in triplicates and repeated at least twice. Final figures have been computed considering all the data sets and results were presented in terms of mean and standard deviation. Data were processed using t-test at p<0.05. Graph Pad Prism (Version 6.01) was used for all the statistical analyses. Median lethal concentra-

**Figure 7. Reactive Oxygen Species profile.** Generation of reactive oxygen species (ROS) was found to increase with time and with concentration suggesting oxidative stress induced toxicity of nanoparticles. (n = 3).

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**Figure 8. Gross internalization of nanoparticles.** Nanoparticles adsorbed onto the animal outer surface and internalized into the gut were found to increase with exposure concentrations. (n = 3).

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Figure 9. Systemic internalization of nanoparticles. Systemic internalization of nanoparticles was seen to remain almost constant till 100 µg/mL nanoparticle concentration after which a slight increase was observed for 120 µg/mL. (n = 3).
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Figure 10. Optical microscopic analysis of *C. dubia* before and after the interaction. (A) Bright-field micrograph shows nearly transparent animal with traces of algal feed in the alimentary canal of the untreated animal. (B) Phase contrast micrograph shows natural coloration of tissues prior to the nanoparticle exposure. (C) Interacted daphnids show deposition of nanoparticle around the alimentary canal. (D) Altered coloration of the tissues can be noted in the phase contrast micrograph of interacted daphnid.
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The colloidal stability of nanoparticles in an aqueous test system is a prime contributory factor to its reactivity thus, leading to toxicity [44]. The size dependent toxic impact of ceria nanoparticles towards *Escherichia coli* has been reported [45]. The correlation between nanoparticle stability and toxicity has also been established in our previous reports, which showed significant toxicity of aluminum oxide nanoparticles towards the freshwater isolate, *Bacillus licheniformis* and *Chlorella sp.* in the initial 24 h when the aggregation effects were less [31], [32]. Another report on the aluminum oxide particles on sediment dwellers revealed significant correlations between the toxic impact and the particles size [46].

Uptake potential is another important aspect determining toxic impact of any nanomaterial. A variation in particle diameter giving rise to a poly-disperse system has a variable uptake frequency for each sized particles [47]. A study with fluorescent labeled polystyrene particles of varying sizes showed 100 nm particles had 2.3 folds greater uptake compared to that of 50 nm particles, 1.3 folds to that of 500 nm particles, about 1.8 folds that of 1000 nm particles. Thus, concluding that nanoparticles of 100–200 nm size acquire the optimized properties for cellular uptake. This effectively counters the misconception of smaller the size, better the uptake potential [47].

In our study, we have had a fairly monodisperse system to start with (0 h) with a polydispersity index of 0.08, but as the time progressed (beyond 24 h) the polydispersity index was measured to be 0.16 and 0.19 at 24 and 72 h respectively, indicating a significant rise in polydispersity.

Now, as concluded by [47], rather than the polydispersity, specific particle size plays major role in the uptake mechanism and the extent of internalization. The particles ranging from 100–200 nm are generally internalized through receptor mediated mechanisms; whereas, larger particles are taken up through phagocytosis [48]. In our study, the gross internalization was predominantly through oral route whereas, systemic internalization is expected to be through receptor mediated pathways during the initial exposures (<24 h) and through phagocytosis beyond 24 h. However, the dynamic state of the experiment made it difficult to comprehend the uptake process specifically.

Bioavailability of Aluminum Oxide Nanoparticles during Exposure

The bioavailability of aluminum oxide nanoparticles was monitored at the stipulated intervals of 24, 48 and 72 h during the exposure assessment. The available concentration of aluminum oxide was found to decrease as both exposure time and initial exposure concentrations increased. At 0 h, the available concentration of aluminum oxide was above 97.66% for administered doses of 20, 40, 60, 80,100 and 120 µg/mL respectively. Subsequently, at 24 h, bioavailable concentrations ranged between 67.66 to 85% for the same dose of administrated concentrations (20, 40, 60, 80,100 and 120 µg/mL) whereas, at 48 h, it was between 48.33 to 65%. Finally at 72 h, 32.66 to 48.5% aluminum oxide was available in the system (Figure 4). The bioavailability of the test substances is a critical element influencing toxicity and is dependent upon factors such as sorption to the vessel wall, sedimentation, aggregation and internalization within the test organism [26]. In this case we may assume that aggregation of the particles in the test medium played a crucial role in defining the bioavailability of aluminum oxide during 72 h exposure. Figure 3 clearly shows a time and dose dependent aggregation profile of the particles which possibly caused settling and consequent decreased bioavailability of the nanoparticles in the test medium. To the best of our knowledge this is the first study
where bioavailability of aluminum oxide particles has been quantified to explain toxic impact in an aqueous matrix.

**Toxicity Analysis**

The immobilization and subsequent mortality of *C. dubia* was dependent on exposure concentrations and time. At 24 h, all the animals were noted to be alive for 20 and 40 μg/mL exposures whereas, 90±3.33, 77±3.33, 57±3.33 and 44.34±3.33% survival was recorded upon exposure to 60, 80, 100 and 120 μg/mL dosages, respectively. At 48 h, 94.34±1.66 and 88±1.66% daphnids were alive upon 20 and 40 μg/mL exposure, whereas 70±5, 96±3.33, 35±5 and 20±5% survival were noted for the 60, 80, 100 and 120 μg/mL doses. Finally at 72 h, 90±6.66, 83.34±3.33, 60±3.33, 45±10, 15±3.33 and 5±1.5% survival was observed for the similar exposure concentrations in the range of 20–120 μg/mL respectively (Figure 5).

Furthermore, the median lethal concentrations were computed to be 117.8 μg/mL, 86.4 μg/mL and 74.3 μg/mL at 24, 48 and 72 h respectively to provide a more conventional measure of toxicity potential of aluminum oxide nanoparticles on *C. dubia*. A clear dose and exposure dependent toxicity profile was noted during the study. The sensitivity towards aluminium oxide and a strong dose and exposure dependence make daphnids a promising candidate for bio indicator in fresh water aquatic systems [30].

A number of reports exist on toxic impacts of different nanomaterials on daphnids. In our recent study on *C. dubia*, median lethal concentrations (*LC₅₀*) of TiO₂ nanoparticles were found to be 8 μg/mL in case of light exposure and 32 μg/mL under dark [33]. The effects of ZnO, CuO and TiO₂ nanomaterials on daphnids. In our recent study on *D. magna*. The silver nanoparticle induced toxicity, uptake, aggregation and bioavailability studies demonstrating a clear decrease in the bioavailability of the nano-sized particles in the test system both with the increasing dose and the exposure time. To clarify this ambiguity, we studied the possible solubilization of aluminium ions from aluminium oxide in the later phase of the test period. Under certain conditions, aluminium oxide nanoparticle has been reported to leach labile aluminium post 24 h exposures [32]. Moreover, ionic aluminium has been reported to impart higher toxicity response compared to aluminium oxide nanoparticles [53]. Based on these assumptions, another assay was taken up to quantify the amount of ionic aluminium at specific time intervals of 24, 48 and 72 h.

**Figure 11. Transmission electron microscopy of the alimentary canal.** Alimentary canal of untreated daphnids shows (A) uniform gut lining with intact microvilli and basal membrane; (B) a closure view of the gut lining showing healthy microvilli and basal membrane. The treated samples show (C) disrupted cellular features with (D) disintegrated microvilli and basal membrane. Certain samples also showed (E) complete destruction of microvilli with few loosely attached cells and (F) abnormal cellular interior. BM: basal membrane; MV: microvilli. The gut lining of at least five different animals were observed to draw a conclusion (n = 5).

**Dissolution of Soluble Aluminium during Exposure**

In this study, the extent of soluble aluminum leached from the nanoparticles has been measured in the test system. During the longer exposures (at 48 h and 72 h), a significant amount of aluminum was found to have leached to the test system. At 48 h, 3.9, 4.1, 4.9, 5.1, 5.6 and 5.2 μg/mL of soluble aluminum was found in the particle free system against the corresponding exposure concentrations of 20, 40, 60, 80, 100 and 120 μg/mL respectively. At 72 h, 4.7, 4.9, 5.7, 5.9, 6.1 and 6.4 μg/mL of soluble aluminum was found in the particle free suspension against the same initial exposure concentrations (Figure 6A). Statistical analysis validated a significant increase in soluble aluminium concentration with respect to exposure time for a particular administered concentration. However, dose dependence was not significant at all exposure concentrations when one step increment was considered.

During the initial exposure period (24 h), the inherent toxic nature of nano-sized particles played a dominant role resulting in a negative impact on the test organisms. Beyond 24 h (i.e. at 48 h and 72 h), labile aluminum leached from aluminium oxide nanoparticle also played a significant role imparting toxic effect. This is substantiated by the dissolution of aluminium ions from the aluminium oxide nanoparticles at later stages as discussed in the preceding section. A similar mechanism was proposed in our previous study with freshwater algae *Chlorella sp* where we showed an increase in the toxicity of aluminium oxide nanoparticles in spite of a significant increase in MHD of particles in the later phases. The probable mechanisms causing dissolution of aluminium oxide particles were discussed extensively and the labile aluminium was found to play the crucial role in toxicity [32]. Another report from our group ruled out any kind of ion mediated toxicity in case of titanium dioxide nanoparticles exposure to *Scenedesmus sp* due to its insoluble nature under the test conditions [33].

The dissolution of the aluminium oxide nanoparticles in aqueous matrix is usually a two-step process involving interaction between surface and matrix leading to dissociation of Al-O complex [54]. During the initial exposure up to 24 h, adsorption of natural organic matter (NOM) onto the nanoparticles due to steric or charge based attractions [55], could have prevented the release of aluminum ions. During the longer exposures of 48 and 72 h, the adsorbed NOM could induce surface reactions resulting in ligand mediated dissolution of Al⁺³ ions from the nanoparticle surface [54], which could be the reason for an increase in the released aluminum ion concentration at 48 h and 72 h.

The soluble aluminium mediated toxicity was quantified using a filtrate containing only the aluminium ions free from the traces of the nanoparticles after 24, 48 and 72 h dissolution. The suspension containing the aluminium ions leached caused no decrease in viability at 24 h even for 120 μg/mL exposure concentrations. A notable drop in the cell viability was noted after 24, 48 and 72 h dissolution. The sensitivity towards aluminium oxide and a strong dose and exposure dependence make daphnids a promising candidate for bio indicator in fresh water aquatic systems [30].

**Oxidative Stress Analysis**

Oxidative stress analysis showed an overall increase in Reactive Oxygen Species (ROS) content with corresponding increment in amount of aluminium oxide nanoparticles and exposure time. At
Gross Internalization of Aluminum Oxide Nanoparticles

A fraction of aluminum oxide nanoparticles have been found to have been internalized into the test organisms after 72 h exposure. The internalized concentration was quantified as $3.7 \pm 0.3$, $3.8 \pm 0.2$, $8.2 \pm 1.4$, $18.3 \pm 2.6$, $21.4 \pm 3.2$, and $23.8 \pm 4.5 \mu g/g$ aluminum oxide metal per gram fresh weight body weight of *C. dubia* at administered particle concentrations of 20, 40, 60, 80, 100 and 120 $\mu g/mL$ respectively (Figure 7). The role of nanomaterials inducing oxidative stress on similar species *Daphnia magna* has been identified as a major contributory factor towards cellular disruption [30]. The pathological conditions due to higher level of oxidative stress in aquatic animals has been detailed by Lushchak et al. [56].

Systemic Uptake of Aluminium Oxide Nanoparticles

After depuration the extent of residual nanoparticle internalized to the tissue of *C. dubia* has been quantified. After 72 h exposure followed by 48 h depuration, $0.9 \pm 0.08$, $0.9 \pm 0.1$, $1.2 \pm 0.09$, $1.2 \pm 0.06$, $1.1 \pm 0.09$ and $1.4 \pm 0.1 \mu g/g$ aluminum oxide per gram body weight of *C. dubia* was found to have been effectively internalized to the tissues for initial administered concentrations of 20, 40, 60, 80, 100 and 120 $\mu g/mL$ respectively (Figure 8). The effective internalization of nanoparticles in the tissues of *Ceriodaphnia* did show an increase with increasing exposure concentrations, a small fraction of the administered dose was found to have deposited inside the body tissues [body burden]. This might have been influenced by several factors like bioavailability, feeding, depuration pattern and disruption of the gut lining. Another major element could be a threshold limit of internalized aluminum oxide nanoparticle beyond which the organisms could not survive.

The stability analysis and the dissolution studies of aluminum oxide nanoparticles in the test medium showed that during initial exposure of 24 h, amount of the leached aluminum ions was insignificant whereas, particulate aluminum oxide was abundant. Contrastingly, during longer exposures (48 h and 72 h) the particulate aluminum oxide was less due to aggregation and subsequent settling of the particles. But there was a significant increase of leached aluminum ions in the system. Based on these observations on abundance of particles/ions, we can assume that during initial exposures the major fraction of the possible uptake would be in the form of particulate aluminum oxide; whereas, late exposures could induce uptake of aluminum ions rather than particulate aluminum oxide. However, on the basis of present analytics, it was not possible to quantify the exact proportion of particle/ion in the internalized aluminum content at any specific time point.

In our previous report, we have estimated the toxicity response of *C. dubia* towards titanium dioxide nanoparticles. A fraction of the administered nanoparticles was found to have internalized into the tissues upon similar treatment of exposure-depuration cycle. The accumulation of the particles in the cellular system contributed towards toxic impact on the organisms [32]. Additionally, the sub lethal accumulation is expected to pass on to the species preying on *C. dubia* upon feeding and will lead to bio-magnification in the food chain and will eventually pose a threat towards the ecosystem [24]. Aluminum oxide showed higher clearance ability and resultant systemic uptake was much lower compared to titanium dioxide retention. This might have resulted in a lesser toxicity response in case of aluminum oxide exposure.

Microscopic Analysis

A preliminary optical microscopy (bright-field and phase contrast) was performed before and after the exposure. Before the exposure, daphnids were found to be nearly transparent with traces of algal feed in the alimentary canal. Contrastingly, after the stipulated exposure period, a large amount of aluminum oxide nanoparticle deposit was noted in and around the mid-alimentary canal (Figure 10).

The alimentary canals of untreated daphnids had regular orientation of microvilli arranged on an intact basal membrane. At lower magnification, the uniformity of the whole gut lining was observed (Figure 11A). A closer view showed intact structural features of microvilli mounted on undamaged basal membrane (Figure 11B). The guts of aluminum oxide nanoparticle treated daphnids were found to have disrupted cellular features throughout the lining (Figure 11C). Higher magnifications showed disintegrated microvilli with basal membrane showing signs of damage (Figure 11D). Another area of gut showed complete destruction of the microvillus lining with a few loosely attached cells (Figure 11E). The outer portion of the basal layer at this area showed signs of disintegration whereas, inner tissues showed several abnormal cellular structures indicating an overall deleterious effect of aluminum oxide nanoparticles (Figure 11F). The findings were comprehensive and similar observations were noted throughout the samples. To ensure statistical validation of the observed features, at least three different sections of the tissues with similar features were noted during analysis. The best observation among those has been included. The findings were in close agreement with our previous study on *Ceriodaphnia* using titanium dioxide nanoparticles [32]. Another study has showed similar observations on *Daphnia magna* upon exposure to copper oxide nanoparticles [49].

Conclusion

The acute exposure study with a strong focus on physicochemical behavior of the particles in the test medium provided a holistic perception of the toxicological response of *Ceriodaphnia dubia* towards the aluminum oxide nanoparticles. The definitive dose response in a dynamic system with time dependent binary contributory factors such as nano-size effect of the particles in the initial phase and contributions from the leached aluminum ions in the later phases was noteworthy. The toxicity data were further substantiated by the evidences of internalization of the particles from the transmission electron microscopic images that provided a clear visualization of the disruptions along the digestive tract indicating a disturbed metabolism inducing cellular death. The results suggest *C. dubia* may be a promising candidate for bio-
monitoring the environmental risk of the nanomaterials in the freshwater ecosystems. Nevertheless, further detailed studies need to be conducted with other environmentally relevant nanomaterials to explore the complex dynamic relationship between the toxicity response and physical-chemical properties of these insoluble tiny particles.

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Author Contributions

Conceived and designed the experiments: SP AM. Performed the experiments: SP SD AH. Analyzed the data: SP SD AH SC AM NC. Contributed reagents/materials/analysis tools: AM. Wrote the paper: SP SD AH SC AM NC.

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