Obtaining modified sorbents based on natural raw materials of Kazakhstan and research of their properties

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ABSTRACT

Kazakhstan takes a leading position in the production of uranium. During the hydrometallurgical processing of uranium-containing raw materials, a significant amount of liquid industrial waste is generated, such as waste solutions that require disposal. One of the most effective methods of cleaning liquid objects contaminated with radionuclides is sorption methods. Synthetic sorbents are not always justified due to their high cost and natural ones due to their low sorption capacity. The production of modified ion-exchange materials based on their combination is an urgent problem in the nuclear industry. The authors considered options for modifying natural aluminosilicate and coal-mineral raw materials of Kazakhstan. For research, zeolite from the previously unexplored Kusmurun deposit and shungite from the Koksu deposit were selected. It is proposed to modify natural sorbents with a tributylphostat and di-2-ethylhexylphosphoric acid mixture in kerosene, a mixture of phosphoric acid and polyacrylamide, technogenic raw materials. The probable mechanism of modification by each of the methods is considered. The sorption properties of the modified sorbents have been studied, and their mechanical strength has been determined.

Keywords: natural sorbents, modification, uranium sorption, mechanical strength.

Introduction

Recently, more and more attention has been paid to the ecological safety of the republic, in connection with the increased threat of environmental pollution by radionuclides, particularly uranium. When processing uranium-containing raw materials, a significant amount of liquid technogenic wastes are formed. The main method of utilization is sorption, which requires the use of inexpensive sorbents of complex action.

Such sorbents can be obtained based on domestic natural raw materials. The significant disadvantages of natural sorbents include low sorption capacity, which can be increased by developing effective and inexpensive methods for their modification.

Various options for obtaining sorbents with improved sorption and kinetic properties are used. Many of them are based on the introduction of additional functional groups into the sorbent structure, which leads to the formation of new
adsorption centers, increasing the sorption capacity and selectivity of the sorbent. For this purpose, use is made of inorganic materials modified with amidoxime or iminodiacetate groups and, salts of heteropoly acids [1, 2]. Sorbents with amidoxime groups on various carriers have shown high efficiency in the extraction of radionuclides, as well as good kinetic properties [3, 4, 5, 6].

To isolate radionuclides from complex technological solutions, sorbents with diphyril, aminophosphinate, carbamoylmethyl-phosphinate, and other phosphorus-containing functional groups have been developed, which can produce stable complexes with radionuclides [7, 8, 9, 10].

The most promising sorbents with functional groups fixed on polymer matrices that form complex compounds are materials obtained on the basis of natural minerals and radionuclide extractants. Such "solid-phase extractants" are characterized by good sorption properties [11].

The synthesis of organopolymers occupies a special place in the production of modified sorbents. This is how an organozelite was synthesized on the basis of natural zeolite-containing tuffs and a water-soluble polymer of polyhexamethylguandine, epichlorohydrin as a cross-linking agent, which simultaneously exhibits cation-exchange, anion-exchange and bactericidal properties [12].

Testing of the sorption capacity of the modified sorbents was carried out under static conditions. Sorption was carried out for 4 hours at room temperature (~ 25°C) at the ratio S:W = 1:5. Desorption was carried out with a solution of 1M sodium carbonate in a static mode at a ratio of S:L = 1:10.

In the course of the research, the mechanical strength of the modified sorbents was also determined in comparison with the initial ion-exchange materials.

In order to determine the effect of activators on mechanical strength, 6 samples were made in the form of pressed briquettes from pre-modified sorbents. Pressed briquettes were made using a PSU-10 hydraulic press designed for static compression testing of standard samples of building materials. The method for modifying natural
sorbents for the manufacture of pressed briquettes, according to the numbering, is presented in Table 1.

**Table 1 - Methods for modifying natural sorbents**

| Sorbent | Method                                                                 |
|---------|------------------------------------------------------------------------|
| Zeolite | 1. Initial                                                              |
|         | 2. Initial Modified with a mixture of di-2-ethylhexylphosphoric acid and tributyl phosphate in kerosene. (Di-2 EGPK + TBP + kerosene) |
|         | 3. Modified with phosphoric acid and polyacrylamide (H₃PO₄ + PAA) |
| Shungite| 4. Initial                                                              |
|         | 5. Modified with a mixture of di-2-ethylhexylphosphoric acid and tributyl phosphate in kerosene. (Di-2 EGPK + TBP + kerosene) |
|         | 6. Modified with phosphoric acid and polyacrylamide (H₃PO₄ + PAA) |

2 series of experiments were carried out. In the first, the samples were pressed at a 200 kg / dm³, water was used as a binder, and briquettes in the form of a cylinder (r = 8, h = 16) were obtained, which were carefully dried and compressed until the first crack. Compression speed 0.1 mm / sec.

In the second, the samples were pressed at 300 kg / dm³, liquid glass was used as a binder, the studies were carried out similarly to the previous one.

**Analysis methods**

The quantitative content of uranium in solutions before and after sorption was determined on an Optima 8000DV inductively coupled plasma atomic emission spectrometer (ICP).

IR spectra were obtained on an Avatar 370 FT-IR spectrometer in the spectral range of 4000-400 cm⁻¹ from preparations in the form of a tablet prepared by pressing 2 mg of a sample and 200 mg of KBr. Experiment attachment: Transmission E.S.P.

To determine the strength of modified sorbents in comparison with the original used a universal floor testing machine AutographAG-X 100 kN, Shimadzu GmbH, Japan.

**The discussion of the results**

On the basis of natural raw materials from Kazakhstan, various options for modifying natural minerals have been developed. [15]. The most promising were the methods that included the treatment of natural minerals with organic extractants: di-2 ethylhexylphosphoric acid in combination with tributyl phosphate, phosphoric acid in combination with polyacrylamide. In the course of the research, we used a previously unexplored zeolite from the Kusmun deposit and shungite from the Koksu deposit after preliminary flotation. We have proposed and tested three options for shungite beneficiation technology. The most promising was the method described in [14]. The product obtained in the flotation process can be classified as shungite concentrate.

IR spectroscopic studies of modified natural sorbents (Figs. 1, 2) showed that when zeolite is modified with a mixture of extractants, the sample contains plagioclase spectra of the albite type Na[AlSi₃O₈] – 762, 747, 726, 647, 590, 528, 465 cm⁻¹. Possibly present: Heulandite Ca[Al₂Si₇O₁₈] ∙ 6 H₂O - 3428, 1032, 523, 465 cm⁻¹, lampentitis Ca[Al₂Si₇O₁₈] ∙ 6 H₂O - 3572, 1032, 762, 523 cm⁻¹, phillipsitis K, Ca[Al₂Si₆O₁₈] ∙ 6H₂O – 3428, 1639, 1032, 590 cm⁻¹, quartz α-SiO₂ – 696, 465 cm⁻¹, calcite CaCO₃ – 1797, 1428, 878, 714 cm⁻¹ [16], di-2 ethylhexyl phosphoric acid (C₁₆H₃₅PO₄) - 2960, 2931, 2874, 2861, 1461, 1384 cm⁻¹ and tributyl phosphate (C₁₂H₂₇PO₄) - 2960, 2873 cm⁻¹ [18].

![Figure 1 – IR spectrum of zeolite modified with a mixture of Di-2 ethylhexylphosphoric acid, tributyl phosphate and kerosene](image-url)
When natural sorbents are modified with a mixture of phosphoric acid and polyacrylamide, for example, shungite, as a result of the interaction of the matrix with modifiers, new compounds are formed, in particular, calcium compounds interact with orthophosphoric acid to form dibasic calcium phosphate. In addition, it can be assumed that when natural sorbents are modified with this mixture, a gel-like film of polyacrylamide is formed on the surface of the matrix, which contributes to an increase in the sorption capacity of sorbents. [22].

Thus, it follows from the data obtained that the mechanism of the formation of modified sorbents based on a matrix of zeolite and shungite by a mixture of extractants di-2 ethylhexylphosphoric acid, tributyl phosphate and kerosene and a mixture of phosphoric acid and polyacrylamide have different nature.

The uranium content in waste solutions, as a rule, is 5-15 mg/dm³. In this regard, we have adjusted the productive solution in accordance with the given uranium concentration and studied the sorption process by modified natural materials. The initial uranium concentration was 11,9 mg/dm³. The kinetic dependences of the sorption of uranium in a static mode from the imitate showed that with all the described modified sorbents it is possible to extract uranium by more than 90% already in the first 45-50 minutes.

In the course of the research, the possibility of repeated use of modified sorbents for the extraction of uranium from liquid radioactive waste was also studied, for which the concentration of di-2-ethylhexylphosphoric acid and tributyl phosphate, as well as phosphoric acid and polyacrylamide, was doubled when modifying zeolite and shungite. Sorption and desorption by modified sorbents were carried out in a static mode, alternating the processes of sorption and desorption. The research results are presented in Table 2.

Table 2 - Results of experiments on sorption and desorption of uranium

| Stage | Process | Zeolite | Shungite |
|-------|---------|---------|----------|
|       | Uranium content, mg/dm³ | The degree of extraction, %, degree of desorption% | Uranium content, mg/dm³ | The degree of extraction, %, degree of desorption% |
| I     | Sorption | 0.79 | 93.36 | 0.046 | 99.6 |
|       | Desorption | 9.50 | 85.51 | 10.62 | 89.6 |
| II    | Sorption | 7.18 | 89.66 | 0.06 | 99.9 |
|       | Desorption | 0.39 | 8.68 | 8.06 | 67.99 |
| III   | Sorption | 7.41 | 37.73 | 0.96 | 91.93 |
|       | Desorption | 0.30 | 6.35 | 3.8 | 32.09 |
It follows from the table that with an increase in the concentration of the modifier, shungite can be used repeatedly. For zeolite, this relationship is not observed.

One of the main factors in sorption is the mechanical strength of the sorbents. We have determined the mechanical strength of the modified sorbents in comparison with the initial ion-exchange materials [23].

The mechanical strength of a material is characterized by its ability to resist various external mechanical influences and is characterized by ultimate strength:

1) when compressed; 2) when stretched; 3) flexural strength and 4) abrasion resistance. We investigated the compressive strength of natural sorbents.

The results obtained are shown in Figure 4, which shows the effect of activators and a binder reagent on the mechanical strength of natural sorbents.

Based on the studies carried out, it can be concluded that the ability of resistance to external mechanical influences (in our case, to compression) during the treatment of zeolite and shungite with phosphoric acid and polyacrylamide increases significantly, and in the case of using a mixture of extractants in kerosene with kerosene it decreases.

It also follows from the figure that the difference in the degree of deformation of the initial sample of zeolite and that modified with a mixture of orthophosphoric acid and polyacrylamide significantly exceeds the similar difference for shungite. At the same time, the decrease in the degree of deformation of a zeolite sample modified with a mixture of extractants in relation to the initial raw material is significantly less than in shungite.

Thus, the optimal options for modifying natural minerals have been determined using the example of zeolite and shungite, and the physicochemical properties of the obtained modified sorbents have been investigated.

It should be noted that the modified sorbents, especially the first two options, have proven themselves well in the testing process. These sorbents can be used for analytical purposes, as well as in low-tonnage production conditions. Their widespread use for the disposal of large volumes of liquid uranium-containing waste is unprofitable. Currently, work in this direction continues. In order to reduce the cost of modified sorbents, studies are being conducted on the possibility of using technogenic raw materials as modifiers, in particular, phosphorus slag, which is a waste of the phosphorus industry and is formed during the electrothermal production of yellow phosphorus. According to the performed physicochemical studies, the main phase of the phosphorus slag - calcium silicate - is represented by the amorphous phase of volostanite. The slag also includes multicomponent glass, small amounts of calcite, ankerite. Phosphorus is present as lazulite.

Analysis of scientific and technical literature in the field of calcium silicate synthesis showed that rational and environmentally friendly options include methods based on the interaction of the initial components in an aqueous medium at elevated temperatures and, in some cases, pressure, i.e. hydrothermal method. The hydrothermal method allows not only the synthesis of hydrosilicates, but also affects their structure and particle morphology. Hydrothermal conditions...
simulate the formation of minerals in the earth’s interior. Calcium carbonate and sodium chloride are commonly used as the aqueous phase.

During research, it was found that during the hydrothermal treatment of slag in a carbonate medium with an increase in temperature, the amorphous phase is transformed into a crystalline phase, and the morphology of particles also changes: the conglomerates existing in the initial sample gradually change their shape and turn into particles of an acicular structure (Fig. 5, 6). During the hydrothermal treatment of phosphorus slag with sodium chloride, its amorphous structure is retained.

Several options for modifying natural minerals have been developed. An indicator of one or another modification method is the sorption process.

The best option for modifying natural minerals (zeolite or shungite) is mixing a natural sorbent with phosphoric slag activated in a chloride or carbonate medium and processing with a polyacrylamide solution.

The structure of the phosphorus slag activated with sodium chloride solution is partially retained even after it has been modified by natural minerals. This is especially clearly seen on the example of shungite (Fig. 7).

**Figure 5 -** The microstructure of the original phosphorus slag with an increase in x3000

**Figure 6 –** Particles of activated (Na₂CO₃ - 150 g / dm³, t - 230 °C) phosphorus slag with an increase x3000

IR spectroscopic analysis of a sample of modified shungite showed the presence of: calcite - CaCO₃ – 1798, 1423, 875, 849, 712 cm⁻¹, quartz - SiO₂ – 1166, 1081, 798, 779, 695, 516, 468, 396, 368 cm⁻¹ [16], [24]. The band at a wavenumber of 319 cm⁻¹ falls within the range of bond manifestations Ca – O [25]. There are: wollastonite - CaSiO₃ – 1081, 1060, 928n, 909n, 516, 468 cm⁻¹ [16], muscovite - KAl₂[(OH, F)₂| AlSi₃O₁₀] – 1028, 928 cm⁻¹ [16], [26], multicomponent glass - 1035, 928 cm⁻¹ [27]. Stretching vibrations of bonds Si–O–Si - 1035 cm⁻¹, shoulder at 928 cm⁻¹ corresponds to stretching vibrations of terminal bonds Si–O–Si [28], bassanite - Ca[SO₄] · ½ H₂O – 634, 602, 468 cm⁻¹ [29].

When activated with sodium carbonate - in the process of modification, the structure is transformed.

Studies have established that the sorption capacity of modified sorbents increases when dressing natural minerals with phosphorus slag. If the slag modified with slag activated in a chloride medium sorbs both uranium and iron, then in a carbonate medium it is mainly iron. This property can be used to separate them.

Thus, the possibility of using technogenic raw materials - phosphorus slags - as modifiers of natural minerals has been shown.
Conclusions

In the course of research, methods have been developed and tested for modifying natural sorbents, which make it possible to actively extract uranium. The modifiers were a mixture of di-2-ethylphosphoric acid and tributyl phosphate in kerosene, phosphoric acid, polyacrylamide and technogenic raw materials. On the basis of IR spectroscopic studies, a prediction was made regarding the mechanism of interaction of modifiers with the matrix of a natural sorbent. It is shown that the mechanism of the formation of modified sorbents based on a matrix of zeolite and shungite with a mixture of extractants di-2-ethylhexylphosphoric acid and tributyl phosphate in kerosene and a mixture of phosphoric acid and polyacrylamide is of a different nature.

The properties of modified sorbents have been studied, and their sorption capacity and mechanical strength have been assessed. It is shown that the degree of uranium extraction by modified sorbents exceeds 90% already in the first 45 - 50 minutes.

It was found that when processing both zeolite and shungite with a mixture of phosphoric acid and an acid and polyacrylamide, the mechanical strength of the sample increases, and a mixture of extractants (di-2ethylhexylphosphoric acid and tributyl phosphate) in kerosene helps to reduce its value.

The possibility of using technogenic raw materials - phosphorus slags - as modifiers of natural minerals is shown.

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Получение модифицированных сорбентов на основе природного сырья
Казахстана и исследование их свойств

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АННОТАЦИЯ
При гидрометаллургической переработке урансодержащего сырья образуется значительное количество жидкотехнических отходов – сбросных растворов, требующих утилизации. Одним из наиболее эффективных приемов очистки загрязненных радионуклидами жидкостей является использование сорбционных методов. Применение синтетических сорбентов не всегда оправдано из-за их высокой стоимости, а природных – из-за низкой сорбционной емкости. Получение модифицированных ионообменных материалов на основе их сочетания является актуальной проблемой атомной промышленности.

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