Simple Approaches for the Synthesis of AgNPs in Solution and Solid Phase Using Modified Methoxypolyethylene Glycol and Evaluation of Their Antimicrobial Activity

Ayman El-Faham 1,2, Hessa H Al-Rasheed 1, Essam N Sholkamy 3, Sameh M Osman 1,4, Zeid A ALOthman 1,4

1 Department of Chemistry, College of Science, King Saud University, Riyadh 11451, Saudi Arabia; 2 Department of Chemistry, Faculty of Science, Alexandria University, Ibrimia, Alexandria 21321, Egypt; 3 Department of Botany and Microbiology, College of Science, King Saud University, Riyadh 11451, Saudi Arabia; 4 Advanced Materials Research Chair, Department of Chemistry, College of Science, King Saud University, Riyadh 11451, Saudi Arabia

Purpose: Simple methodology for preparation of metal nanoparticles such as AgNPs uses an methanolic aqueous medium at room temperature or a solvent-free procedure under microwave irradiation. The prepared AgNPs showed a significant antimicrobial effect against Gram-positive bacteria, Gram-negative bacteria, and fungi.

Methods: The modified methoxypolyethylene glycol bishydrazino-s-triazine (mPEGTH2) showed remarkable activity for reducing Ag+ to Ag0 in an aqueous methanolic solution and using a solvent-free method (solid phase) under microwave irradiation. In the solid phase synthesis, the size and shape of the AgNPs can be controlled by varying the weight ratio of mPEGTH2 to AgNO3 used. In addition, the antimicrobial activity depends on the ratio of mPEGTH2 to AgNO3. The mPEGTH2-AgNPs (2:1) demonstrated higher antimicrobial activity compared to mPEGTH2-AgNPs (1:1) against Gram-positive bacteria, Gram-negative bacteria, and C. albicans.

Results: This work presents simple methods for the synthesis of AgNPs using modified methoxypolyethylene glycol with bishydrazino-s-triazine (mPEGTH2); a solution method, using methanol-water medium at room temperature, and a solvent-free (solid phase) method, employing microwave irradiation or direct heating which could be used for the preparation of AgNPs on large scale. In the solid phase, ratios of mPEGTH2 to AgNO3 (1:1 or 2:1, respectively) are very important to control the size and shape of AgNPs. While in solution phase is not necessary where the molar ratio used is 10:1. Most of the experimental methods resulted in AgNPs ranging in size from 7 to 10 nm as observed from XRD and TEM characterization. The antimicrobial activity of the AgNPs was also dependent on the weight ratio of mPEGTH2 to AgNO3, with a large effect as observed when using the solvent-free method. The mPEGTH2-AgNPs (2:1) demonstrated higher antimicrobial activities compared to mPEGTH2-AgNPs (1:1) against S. aureus, S. epidermidis, E. faecalis, E. coli, P. aeruginosa, S. typhimurium, and C. albicans. In all cases, the MICs and MBCs of mPEGTH2-AgNPs (1:1) were lower than those of mPEGTH2-AgNPs (2:1).

Conclusion: In summary, mPEGTH2-AgNPs (2:1) is a promising candidate to kill pathogenic microbes. In particular, the method used for the preparation of AgNPs by using polyethylene glycol polymer modified with bishydrazino-s-triazine has the most potential and would be the most cost-effective method. This method of the synthesis of nanoparticles may be suitable for the preparation of other metal nanoparticles, which would allow for numerous applications in medicinal and industrial.

Keywords: methoxypolyethylene glycol bishydrazino-s-triazine, silver nanoparticles, solution phase, solid phase, antimicrobial activity
Introduction

Nanoparticles have important roles in materials science, chemistry, physics, medicine, and numerous other fields.1–4 Therefore, there is a strong need to create a new methodology for the synthesis of metal nanoparticles in large scale. Silver nanoparticles are utilized with increasing frequency in various fields, including medicine and agriculture, due to their unique physical and chemical properties.5 Most notably, AgNPs have been used for pharmaceutical applications, such as antibacterial6 agents and anticancer agents.7,8 Recently, the use of AgNPs has expanded to other applications, such as textiles, wound dressings, and biomedical devices.9–11

Various methods have been used for the preparation of AgNPs. Generally, the synthesis of nanoparticles has been carried out using three different approaches: Physical methods,12,13 chemical methods using water or organic solvents,14–16 and biological methods.17–19

Chemical methods usually employ three main components, which are metal precursors, reducing agents, and stabilizing/capping agents. The major advantage of chemical methods is that they have high yields compared to that of physical methods. On the other hand, chemical methods are tremendously expensive; additionally, the materials used for AgNPs synthesis, such as citrate, borohydride, thioglycerol, and 2-mercaptoethanol are toxic and hazardous materials. Furthermore, during the synthetic process, numerous toxic and hazardous byproducts are generated.20

Unlike chemical methods used for AgNPs synthesis, biological methods21 are simple, rapid, non-toxic, dependable, and green; additionally, the AgNPs produced are of well-defined size and morphology under optimized conditions. However, particle characterization is necessary because the physicochemical properties of the particles could have a significant impact on their biological properties.22,23

Several methods have been presented for the synthesis of AgNPs using green, cost effective, and biocompatible methods without the use of toxic chemicals. In this green chemistry approach, researchers have used several different types of bacteria.24–26 Additionally, several biomolecules, such as biopolymers,27 starch,28 and amino acids29 have been used. The use of biological molecules for the synthesis of AgNPs is eco-friendly and environmentally safe, and appeared to be useful for controlling of particle size and shape, which is an important factor in biomedical applications.30,31

Poly(ethylene glycol) (PEG) is a biocompatible polymer used extensively in many fields, including drug delivery, gene delivery, lithium polymer electricity storage systems, etc.32 Two approaches have been previously used to activate the hydroxyl terminal of PEG through the changing the hydroxyl group to a more reactive group.33

In addition, methoxypolyethylene glycol (mPEG) is a mono-protected form of PEG that is a biocompatible polymer and has unique properties that make it particularly useful for many pharmaceutical applications.34 The unique solubility behavior of mPEG in water and several organic solvents allowed mPEG and its derivatives to be easily to purify from low molecular weight organic impurities by using simple methods, such as precipitation and centrifugation. However, the hydroxyl group of mPEG is not reactive enough to allow direct reaction with other molecules. Thus, the terminal hydroxyl group of the mPEG must be derivatized to a more reactive group to allow for reaction with other groups present in the biomolecule. Cyanuric chloride (2,4,6-trichloro-s-triazine) is typically used to activate the terminal hydroxyl group of mPEG due to its low cost, commercially availability, chemo-selective reactivity, and biocompatibility due to its having three reactive sites.35–38

In this regard, we report here the synthesis of an mPEG bis-hydrazino-s-triazine derivative and its use as a reducing agent to produce AgNPs using solution phase (methanol-water) or solvent-free method (solid phase) under microwave irradiation, the antimicrobial activity of the synthesized AgNPs was evaluated as well.

Materials and Methods

Methoxypolyethylene glycol (mPEG) with molecular weight 2000 or 5000 and 2,4,6-trichloro-s-triazine (cyanuric chloride) were purchased from Sigma-Aldrich. The AgNO₃ salt was purchased from Merck. All of the chemicals used were analytical grade and were used without further purification. A Fourier transform infrared (FT-IR) spectrometer (Nicolet, NEXUS-670); a UV-vis spectrophotometer (Shimadzu UV-1208 model); and an X-ray diffractometer (XRD) (X’Pert PRO, PANalytical BV, Almelo, the Netherlands) using CuKα radiation were used for the characterization of AgNPs.39 Transmission electron microscopy (TEM), equipped with energy dispersive X-ray (EDX) analysis, was used to study the size, morphology, and composition of the AgNPs. Histograms of the size distribution of the AgNPs were calculated from
the TEM images by measuring the diameters of at least 50 particles.39

**Synthesis of Methoxypolyethylene Glycol Dichloro-s-Triazine (mPEGTC2)**

Methoxypolyethylene glycol (mPEG, 20 g of 2000 M.wt. or 50 g of 5000 M.wt.) was dissolved in anhydrous benzene (200 mL), and then cooled to a temperature between 5°C and 10°C. Cyanuric chloride (fivefold molar excess, 9.2 g in 200 mL benzene) was slowly added to the solution of mPEG, followed by the addition of anhydrous sodium carbonate (5.25 g, fivefold molar excess). The reaction mixture was stirred at room temperature for 24h, and then the solid inorganic salt was filtered out of solution. Excess petroleum ether 40/60 was added to the filtrate and mPEGTC2 derivative was precipitated as a white solid. The precipitate was collected, then dissolved in benzene (200 mL) and precipitated again with petroleum ether 40/60. This process was repeated twice to remove any unreacted cyanuric chloride and afford the pure intermediate product in 82–84% yield.

**Synthesis of Methoxypolyethylene Glycol Bishydrazino-s-Triazine (mPEGTH2)**

The synthesized mPEGTC2 (10 g) was suspended in 200 mL of ethanol, followed by the slow addition of 20 mL of hydrazine hydrate (80%) under stirring at room temperature. The reaction mixture was refluxed for 6 to 8 h, and then cooled to room temperature; the excess solvent and hydrazine hydrate were removed under vacuum. The crude product was treated with excess ether to afford a white solid, which was collected by filtration, washed with ether, and then dried under vacuum to produce the final product, mPEGTH2, in nearly quantitative yield.

**Synthesis of Silver Nanoparticles**

**Liquid Phase Synthesis of AgNPs at Room Temperature**

A solution of 10⁻³ M AgNO₃ in 20 mL of deionized water was added to a suspended solution of 10⁻² M mPEGTH2 in 20 mL methanol with stirring at room temperature. Stirring of the reaction mixture was continued for 24h. The color of the solution changed from colorless to yellowish-red then to black (Figure S1). The AgNPs were collected using an ultracentrifuge (8000 rpm), washed with aqueous methanol, and then dried under vacuum.

**Solid Phase (Solvent-Free) Synthesis of AgNPs Using Conventional Heating**

AgNO₃ salt was mixed with mPEGTH2 in two weight ratios (1:1 and 1:2, respectively) and then heated directly at 80°C for 30-sec (at 5-sec intervals) with continuous mixing; during heating, the color changed from white to grey and then to dark grey. The mixed solid product was washed with aqueous methanol, then collected using ultracentrifuge, and then dried under vacuum for 24h.

**Solid Phase (Solvent-Free) Synthesis of AgNPs Using Microwave Irradiation**

The AgNO₃ was mixed well with mPEGTH2 in two different weight ratios (1:1 and 1:2, respectively) at room temperature. The solid mixture was heated using microwave irradiation at 60°C, 600 W for 5-sec to 10-sec. The AgNPs were suspended in a 1:1 methanol-water mixture, collected using an ultracentrifuge, washed with aqueous methanol, and then dried under vacuum.

**Biology**

**Test-Microorganism**

The antibacterial activities of mPEGTH2Ag (2:1) and mPEGTH2Ag (1:1) were assessed against two bacterial groups and one fungus. Gram-positive bacteria (Staphylococcus aureus ATCC 29213, Staphylococcus epidermidis ATCC 12228, and Enterococcus faecalis ATCC 29212) and Gram-negative bacteria (Escherichia coli ATCC 25922, Pseudomonas aeruginosa ATCC 27853, and Salmonella typhimurium ATCC 14028) were maintained in a Brain Heart Infusion (BHI) medium at 20°C; 300 mL of each stock culture was added to 3 mL of BHI broth. Overnight cultures were kept for 24h at 37 ± 1°C. After 24h of incubation, the purity was checked, and the bacterial suspension (inoculum) was diluted with a sterile physiological solution (for the diffusion and indirect bioautographic tests) to 10⁸ CFU/mL (turbidity = McFarland barium sulfate standard 0.5). In the case of the fungus Candida albicans ATCC 60193, the medium used for antagonistic activity against tested fungi was Potato Dextrose Agar (PDA).

**Antimicrobial Activity Evaluation (Well Diffusion Method)**

The samples of mPEGTH2Ag (2:1 and 1:1) were prepared at a concentration of 3 mg/mL, dissolved in dimethylsulfoxide (DMSO). Preparation of sterilized Mueller-Hinton agar plates seeded with pathogenic bacteria was performed; 60 µL of each ratio of mPEGTH2Ag was
prepared and placed in wells on the agar plates. The bacterial plates were incubated at 37°C for 24h, while in case of fungi; the used medium in antagonistic activity against tested fungi. Antimicrobial activities were determined by the inhibition zone after incubation. Discs of gentamicin (30 µg) was used as a positive control in all of the experiments.

**Minimum Inhibitory Concentration (MIC) Determination**

The antibacterial activity of mPEGTH2Ag (2:1) and mPEGTH2Ag (1:1) was studied by employing a micro-dilution method, using Mueller-Hinton broth according to the clinical and laboratory standards institute (CLSI) guideline. The inoculum was prepared as described previously. Stock concentrations of mPEGTH2Ag (2:1) and mPEGTH2Ag (1:1) were both 3 mg/mL. Further 1:2 serial dilutions were performed by addition of the culture broth to reach concentrations ranging from 1.5 to 0.235 mg/mL. Then, 90 µL samples of the different dilutions were distributed in separate wells of a 96-well plate, as well as a sterility control and a growth control (containing culture broth plus DMSO, without antimicrobial substance). Each test and growth control well was inoculated with 10 µL of a bacterial suspension (10^8 CFU/mL). All experiments were performed in triplicate and the micro-dilution trays were incubated at 37 ºC for 24 h. After incubation, the culture broth of each dilution was inoculated on nutrient agar to detect MIC and MBC.

**Results and Discussion**

**Synthesis of mPEGTH2**

The chemical reduction method is the most commonly used method for the production of silver nanoparticles (AgNPs) despite the fact that it requires toxic and hazardous inputs and produces environmentally harmful byproducts. In order to avoid using and producing toxic substances, many green methods have been reported.

In this respect and in our ongoing research for the synthesis of metal nanoparticles (MNPs), we report here that the modification of mPEG to mPEG-bishydrazino-s-triazine derivative (mPEGTH2) and its application to the synthesis of AgNPs by employing methanolic aqueous medium or solvent-free conditions under microwave irradiation or conventional heating to provide a suitable alternative to traditional, chemical-based methods for the synthesis of AgNPs.

In the first step of this method, mPEG was activated using cyanuric chloride 1 to afford the dichloro-s-triazine derivative mPEGTC2 2, and then reacted with hydrazine hydrate to afford the bishydrazino-s-triazine derivative mPEGTH2 3 in approximately quantitative yield (Scheme 1).

The FTIR of mPEGTH2 showed two sharp peaks at 3540 and 3420 cm⁻¹ related to the NH (NH stretching vibration), 2860 showed weak –C-H stretch, and 1560 cm⁻¹ (NH bending vibration), and 1097 cm⁻¹ (C-O stretching vibration) (Figure S2).

**Synthesis of AgNPs**

The first approach was using an aqueous solution method in which a solution of AgNO₃ was added to a methanolic solution of mPEGTH2 at room temperature and the mixture was gently stirred overnight at room temperature to reduce the Ag⁺ to Ag⁰. The color of the solution changed from colorless, to yellow, and then to red (Figure S1). The AgNPs were collected using an ultracentrifuge for 20 min (8000 rpm), dried, and then characterized using several techniques. The proposed mechanism for reduction of Ag⁺ to Ag⁰ is illustrated in Scheme 2. From this proposed mechanism, it is implied that the presence of van der Waals forces between the positive charge of Ag⁺ and the oxygen of the mPEGTH2 increase the capping efficiency of mPEGTH2 for the surface of the AgNPs. In the second approach, both the microwave irradiation and the conventional heating assisted the reduction of Ag⁺ to Ag⁰ via dimerization. This is likely due to the aggregation of Ag clusters into nanoparticles (Scheme 2).
addition, increasing the weight ratio of mPEGTH2 to AgNO$_3$ (from 1:1 to 2:1, respectively, for the solvent-free method), is recommended to enhance the reduction step and ensure the capping of AgNPs. In addition, increasing the ratio of mPEGTH2 would also help to control the particle size of AgNPs and increase their stability for several months in typical environmental conditions.

**Characterization of AgNPs**

**FTIR and UV-Spectroscopy Characterization**

The FTIR spectrum for the mPEGTH2-AgNPs (Figure S3) was analyzed to identify the active group of the mPEGTH2, which are responsible for the reduction of the Ag$^+$ to Ag$^0$. The FTIR results of mPEGTH2-AgNPs showed broad absorption peak at 3450 cm$^{-1}$ related to the NH (stretching vibration) of the hydrazino groups, 2885 weak –C-H stretch, and peak at 1467 cm$^{-1}$ for the NH (bending vibration), 1360 cm$^{-1}$ represented the CH$_2$ bending and the 1109 cm$^{-1}$ represented the C-O stretching vibration.

The broadness of the peaks at 3444 cm$^{-1}$ for the NH groups, confirmed that the hydrazine group responsible for the reduction of Ag$^+$ to Ag$^0$. The obtained data agreed with the proposed mechanism shown in Scheme 2 for the formation of AgNPs.

The UV-visible spectroscopy (Figure 1) also confirmed the formation of AgNPs in the presence of mPEGTH2. A strong and slightly broadband for mPEGTH2Ag was observed at 424 nm. This band is correlated to the surface plasmon absorption band, which explains the formation of the uniform, dispersed spherical AgNPs capped with mPEGTH2.
mPEGTH2. The broadness of the band decreased in intensity, with appearance of a shoulder at 375 nm, for mPEGTH2-AgNPs (AgNO₃:mPEGTH2, 1:1) as shown in Figure 1A. This observation indicates an accumulation of AgNPs with the change in chemical structure of mPEGTH2. The obtained data agreed with the proposed mechanism shown in Scheme 2 for the formation of AgNPs, confirming a dependence on the weight ratio of mPEGTH2 to AgNO₃. This ratio is necessary in solid phase synthesis to control the size and shape of the AgNPs. While in case of the solution phase method, it is not necessary because the molar ratio used is 10:1 mPEGTH2:AgNO₃.

**EDX and XRD Analysis**

The elemental analysis of the AgNPs was performed using EDX analysis, as represented in Figure 2; the EDX analysis confirmed the presence of the silver signal related to the AgNPs. Signal peaks were observed in the range of 2.5–4.0 keV, which corresponds to the binding energies of crystalline silver. The strong signal peak near 0.2 keV corresponds to the carbon in the mPEGTH2, connected to the AgNPs. These results were confirmed by previous reports that showed that AgNPs are crystalline in nature, as supported by the same EDX results.

The crystal structure of the mPEGTH2-Ag nanoparticles was determined from the XRD patterns, as represented in Figure 3 (solution and solid-phase method). The data show a broad peak at 2θ = 20°, confirming the amorphous structure of mPEGTH2. The mPEGTH2-AgNPs sample showed peaks at about 38.1°, 44.09°, 64.36°, 77.29°, and 81.31° for both of the prepared materials, which related to the 111, 200, 220, 311, 222, 400, 331, and 420 planes, respectively. The intensive diffraction peak at 38.1°, which is related to the {111} lattice plane of face-centered cubic silver, indicates that pure AgNPs were formed. The diffraction planes of the mPEGTH2-AgNPs indicated the presence of a typical face-centered cubic structure of silver, according to the available literature (Joint Committee on Powder Diffraction Standards, JCPDS file No 04–0783).

**TEM Analysis**

The uniform and well-dispersed spherical shape of the produced AgNPs and the resulting histogram were observed from the TEM images (Figure 4A–C). The uniform shape is caused by the good coordinating and capping capability of the N-atom in mPEGTH2 to reduce the Ag⁺ to Ag⁰.

The Average particle sizes (D) were calculated using the following equation:

$$D = \frac{0.9 \lambda}{\beta \cos \theta}$$

where β is the full-width at half-maximum of peaks and λ is the X-ray wavelength.

The data showed that the particle sizes of mPEGTH2-AgNPs synthesized using the solvent-free and the solution phase methods (Figure 5), showed the formation of AgNPs with diameter of 7 to 10 nm; this confirms that the weight ratio of mPEGTH2 related to the AgNO₃ is important in coating all silver particles, especially in the solvent-free method.

**Antimicrobial Activity**

The antimicrobial activities for the prepared mPEGTH2-Ag nanoparticles were evaluated against selected microorganisms.
The results indicated that mPEGPTH2-AgNPs obtained from the weight ratio 2:1 (mPEGTH2:AgNO₃) exhibited a wide range of antimicrobial activity against tested Gram-positive and Gram-negative bacteria, in addition to the inhibition of *C. albicans*. In addition, they are more effective than mPEGPTH2-AgNPs obtained from weight ratio 1:1 (mPEGTH2:AgNO₃) (Table 1, Figure 6). The results obtained in this study agree with the reported results by Abul et al, who reported that MK-AgNPs as an effective antibacterial agent against Gram-positive and Gram-negative bacteria.⁴⁹

By the typical duplicate two-fold using dilution method, MICs and MBCs were determined for mPEGTH2-AgNPs (2:1) and mPEGTH2-AgNPs (1:1) against *E. coli, S. epidermidis,* and *C. albicans*. The results represented in Table 2 showed stronger bioactivity of mPEGTH2-AgNPs (2:1) compared to the bioactivity of mPEGTH2-AgNPs (1:1) against the tested pathogenic microbes. The results also revealed that mPEGTH2Ag (2:1) was a strong antifungal agent against *C. albicans* and a strong antibacterial agent against *S. epidermidis*. The obtained results agree well with the previously reported results for polyethylene glycol coated nanoparticles that were effective in killing all bacterial strains, including *E. coli DH5α, Bacillus subtilis, Micrococcus luteus,* and *S. aureus*.⁵⁰,⁵¹ The results also agreed with the reported results by Popa et al; who reported that polyethylene glycol coated silver nanoparticles most stable and effective.⁵²

**Conclusion**

This work represents simple and efficient methods for the synthesis of AgNPs using modified methoxypolyethylene glycol with bishydrazino-s-triazine mPEGTH2. The AgNPs were prepared using two methods; a solution method, using methanol-water medium at room temperature, and a solvent-free (solid phase) method, employing

---

**Table 1** Determination of Microbial Growth Inhibition Diameters (Mm) by mPEGTH2Ag 2:1 and mPEGTH2Ag (1:1)*

| Tested Compd         | *S. aureus* | *S. epidermidis* | *E. faecalis* | *E. coli* | *S. typhi* | *P. aeruginosa* | *C. albicans* |
|----------------------|-------------|------------------|---------------|-----------|------------|-----------------|--------------|
| mPEGTH2Ag (1:1)     | 10          | 15               | 0             | 11        | 0          | 10              | 11           |
| mPEGTH2Ag (2:1)     | 11          | 18               | 18            | 20        | 14         | 17              | 13           |
| Gentamicin**         | 25          | 23               | 17            | 22        | 20         | 19              | —            |

Notes: *DMSO was used as a negative control; **30 µg per disc.
microwave irradiation or direct heating. In aqueous solution, the ratio was 10:1 mPEGTH2:AgNO₃, while in the solid phase, ratios of 1:1 or 2:1, respectively (wt./wt.) were used. Most of the experimental methods resulted in AgNPs ranging in size from 7 to 10 nm. The formation of AgNPs was dependent on the weight ratio of mPEGTH2 to AgNO₃, which allows for control of the size and shape of the AgNPs, especially when using the solvent-free method. The antimicrobial activity of the AgNPs was also dependent on the ratio of mPEGTH2 to AgNO₃, with a large effect observed when using the solvent-free method. The mPEGTH2-AgNPs (2:1) demonstrated higher antimicrobial activity compared to mPEG-TH2Ag 1:1 against S. aureus, S. epidermidis, E. faecalis, E. coli, P. aeruginosa, S. typhimurium, and C. albicans. In all cases, the MICs and MBCs of mPEGTH2-AgNPs (1:1) were lower than those of mPEGTH2-AgNPs (2:1).

In summary, mPEGTH2-AgNPs (2:1) is a promising candidate to kill pathogenic microbes; in particular, the AgNPs synthesized using methoxypolyethylene glycol polymer modified by coating with bishydrazino-s-triazine have the most potential and would be the most cost-effective to produce. This method reported here for the synthesis of silver nanoparticles may be suitable for preparation AgNPs on large scale and could be applicable for other metal nanoparticles, which would allow them for numerous applications in medicine and industry.

Acknowledgments
The authors extend their appreciation to the Researchers Supporting Project at King Saud University, Riyadh, Saudi Arabia, for funding this work through Project No. (RSP-2019/50).

Author Contributions
All authors contributed to data analysis, drafting or revising the article, gave final approval of the version to be published, and agree to be accountable for all aspects of the work.

Disclosure
Professor Ayman El-Faham reports grants from Researchers Supporting Project at King Saud University, Riyadh, Saudi Arabia, Project No. (RSP-2019/50), outside

Table 2 Minimum Inhibitory (MIC) and Minimum Bactericidal Concentration (MBC) of mPEGTH2Ag (1:1) and mPEGTH2Ag (2:1) Against S. epidermidis, E. coli and C. albicans Growth

| Compd. Microbes | mPEGTH2-Ag (1:1) | mPEGTH2-Ag (2:1) |
|-----------------|------------------|------------------|
|                 | MIC (mg/mL) | MBC (mg/mL) | MIC (mg/mL) | MBC (mg/mL) |
| E. coli         | 0.375    | 0.563     | 0.282    | 0.375     |
| S. epidermidis  | 0.375    | 0.563     | 0.094    | 0.188     |
| C. albicans     | 0.282    | 0.375     | 0.036    | 0.047     |

Figure 6 Growth inhibition images of target pathogenic microbes with (A) mPEGTH2-Ag NPs (1:1 ratio); (B) mPEGTH2-Ag NPs (2:1 ratio); (C) negative control DMSO alone.
the submitted work. In addition, Professor Ayman El-Faham has a patent issued: US Patent 9873153 B1. The authors report no other conflicts of interest in this work.

References

1. Zhang XF, Liu ZG, Shen W, Gurunathan S. Silver nanoparticles: synthesis, characterization, properties, applications, and therapeutic approaches. Int J Mol Sci. 2016;17(9):1534. doi:10.3390/ijms17091534
2. Fathima JB, Pugazhendhi A, Venis R. Synthesis and characterization of ZrO2 nanoparticles-antimicrobial activity and their prospective role in dental care. Microb Pathog. 2017;110:245–251. doi:10.1016/j.micpath.2017.06.039
3. Pugazhendhia A, Kumar SS, Manikandan M, Saravanan M. Photocatalytic properties and antimicrobial efficacy of Fe doped CuO nanoparticles against the pathogenic bacteria and fungi. Microb Pathog. 2018;122:84–89. doi:10.1016/j.micropath.2018.06.016
4. Sin W, Barnard RT, Blaskovich MAT, Zoria ZM. Antimicrobial silver in medical and consumer applications: a patent review of the past decade (2007–2017). Antibiotics. 2018;7:93. doi:10.3390/antibiotics7040093
5. Shanmuganathan R, Karuppusamy I, Saravanan M, et al. Synthesis of silver nanoparticles and their biomedical applications - a comprehensive review. Curr Pharm Des. 2019;25:2650–2660. doi:10.2174/138112828566190708185506
6. Saravanan M, Arokayaraj S, Lakshmi T, Pugazhendhi A. Synthesis of silver nanoparticles from Phenerochaeta chrysosporium (MTCC-787) and their antibacterial activity against human pathogenic bacteria. Microb Pathog. 2018;117:68–72. doi:10.1016/j.micropath.2018.02.008
7. Pugazhendhi A, Edison TNJI, Karuppusamy I, Kathirvel B. Inorganic nanoparticles: a potential cancer therapy for human welfare. Int J Pharm. 2018;539:104–111. doi:10.1016/j.ijpharm.2018.01.034
8. Gurunathan S, Park JH, Han JW, Kim JH. Comparative assessment of the apoptotic potential of silver nanoparticles synthesized by Bacillus tequilensis and Calcoybe indica in MDA-MB-231 human breast cancer cells: targeting f53 for anticancer therapy. Int J Nanomed. 2015;10:4203–4222. doi:10.2147/IJN.S83953
9. Li CY, Zhang YJ, Wang M, et al. In vivo real-time visualization of tissue blood flow and angiogenesis using Au,S quantum dots in the NIR-II window. Biomaterials. 2014;35:393–400. doi:10.1016/j.biomaterials.2013.10.010
10. Shanmuganathan R, MubarakAidi D, Prabakar D, et al. An enhancement of antimicrobial efficacy of biogenic and ceftriaxone-conjugated silver nanoparticles: green approach. Environ Sci Pollut Res. 2018;25(11):10362–10370. doi:10.1007/s11356-017-9367-9
11. Möhler JS, Sinne W, Blaskovich MAT, Cooper MA, Zoria ZM. Silver bullets: a new lustre on an old antimicrobial agent. Biotech Adv. 2018;36:1391–1411. doi:10.1016/j.biotechadv.2018.05.004
12. Kruis FE, Fissan H, Rellinghaus B. Sintering and characterization of nanomaterials. Biotechnol Adv. 2014;32:711–726. doi:10.1016/j.biotechadv.2013.11.006
13. Lin PC, Lin S, Wang PC, Sridhar R. Techniques for physicochemical characterization of nanomaterials. Biotechnol Adv. 2014;32:711–726. doi:10.1016/j.biotechadv.2013.11.006
14. Gurunathan S, Han JW, Kwon DN, Kim JH. Enhanced antibacterial and anti-biofilm activities of silver nanoparticles against Gram-negative and Gram-positive bacteria. Nanoscale Res Lett. 2014;9:373. doi:10.1186/1556-276X-9-373
15. Malllick K, Witcomb MJ, Scurrell MS. Polymer stabilized silver nanoparticles: a photocatalytic synthesis route. J Mater Sci. 2004;39:4459–4463. doi:10.1023/B:JMSC.0000034138.80116.50
16. Pugazhendhi A, Prabakar D, Jacob JM, Karuppusamy I, Saratalle RG. Synthesis and characterization of silver nanoparticles using Gelidium amansii and its antimicrobial property against various pathogenic bacteria. Microb Pathog. 2018;114:41–45. doi:10.1016/j.micpath.2017.11.013
17. Qi L, Liu Z, Wang N, Hu Y. Facile and efficient in situ synthesis of silver nanoparticles on diverse filtration membrane surfaces for antimicrobial performance. Appl Surf Sci. 2018;456:95–103. doi:10.1016/j.apsusc.2018.06.066
18. Saravanan M, Barik SK, MubarakAidi D, Prakash P, Pugazhendhi A. Synthesis of silver nanoparticles from Bacillus brevis (NCIM 2533) and their antibacterial activity against pathogenic bacteria. Microb Pathog. 2018;116:221–226. doi:10.1016/j.micpath.2018.01.038
19. Samuel MS, Jose S, Selvarajan E, Mathimani T, Pugazhendhi A. Biosynthesized silver nanoparticles using Bacillus amyloliquefaciens: application for cytotoxicity effect on A549 cell line and photocatalytic degradation of p-nitrophenol. J Photochem Photobiol B. 2020;211:111643. doi:10.1016/j.jphotobiol.2019.111642
20. Murdock RC, Brayditch-Stolle L, Schrand AM, Schlager JJ, Hussain SM. Characterization of nanomaterial dispersion in solution prior to in vitro exposure using dynamic light scattering technique. Toxicol Sci. 2008;101(2):239–253. doi:10.1093/toxsci/fkm240
21. Jacob JM, John MS, Jacob A, et al. Bacterial coating of paper towels via sustainable biosynthesis of silver nanoparticles using Ocinum sanctum leaf extract. Mater Res Express. 2019;6(4):045401. doi:10.1088/2053-1591/aafed
22. Jo DH, Kim JH, Lee TG, Kim JH. Size, surface charge, and shape determine therapeutic effects of nanoparticles on brain and retinal diseases. Nanomedicine. 2015;11:1603–1611.
23. Albanean A, Tang PS, Chan WC. The effect of nanoparticle size, shape, and surface chemistry on biological systems. Ann Rev Biomed Eng. 2012;14(1):1–16. doi:10.1146/annurev-bioeng-071811-071014
24. Gour A, Kumar NJ. Advances in green synthesis of nanoparticles. Artif Cells Nanomed Biotechnol. 2019;47(1):844–851. doi:10.1080/21691401.2019.1577878
25. Gurunathan S, Han JW, Dayem AA, et al. Green synthesis of anisotropic silver nanoparticles and its potential cytotoxicity in human breast cancer cells (MCF-7). J Ind Eng Chem. 2013;19:1600–1605.
26. Gurunathan S, Jeong JK, Han JW, Zhang XF, Park JH, Kim JH. Multidimensional effects of biologically synthesized silver nanoparticles in Helicobacter pylori, Helicobacter felis, and human lung (L132) and lung carcinoma A549 cells. Nanoscale Res Lett. 2015;10:1–17. doi:10.1186/s11671-015-0747-0
27. Gurunathan S. Biologically synthesized silver nanoparticles enhances antibiotic activity against Gram-negative bacteria. J Ind Eng Chem. 2015;29:217–226. doi:10.1016/j.jiec.2015.04.005
28. Leung TC, Wong CK, Xie Y. Green synthesis of silver nanoparticles using biopolymers, carboxymethylated-curdlan and fucoidan. Mater Chem Phys. 2010;121(3):402–405. doi:10.1016/j.matchemphys.2010.02.026
29. Kumar B, Smita K, Cumbal L, Debut A, Pathak RN. Sonoechemical synthesis of silver nanoparticles using starch: a comparison. Bioinorg Chem Appl. 2014;2014:784268. doi:10.1155/2014/784268
30. Shankar S, Rhim JW. Amino acid mediated synthesis of silver nanoparticles and preparation of antimicrobial agar/silver nanoparticles composite films. Carbohydr Polym. 2015;130:353–363. doi:10.1016/j.carbpol.2015.05.018
31. Suganya M, Gnanamangai BM, Govindasamy C, et al. Mitochondrial dysfunction mediated apoptosis of HT-29 cells through CS-PAMAM dendrimers and polyethylene glycols. Eur Polym J. 1983;19(12):1177–1183. doi:10.1016/0014-851. doi:10.1080/0014-851.1983.1000162-2
32. Namazi H, Adeli M. Solution proprieties of dendritic triazine/poly (ethylene glycol) adversely affects PEG-Asparaginase therapy in acute Lymphoblastic Leukemia patients. Cancer. 2007;110(1):103–111. doi:10.1002/(ISSN)1097-0142
35. Sohrabi N, Rasouli N, Torkzadeh M. Enhanced stability and catalytic activity of immobilized α-amylase on modified Fe3O4 nanoparticles. Chem Eng J. 2014;240:426–433. doi:10.1016/j.cej.2013.11.059
36. González M, Vaillard SE. Evolution of reactive mPEG polymers for the conjugation of peptides and proteins. Curr Org Chem. 2013;17:975–998. doi:10.2174/1385272811317090010
37. Odajima T, Onishi M. A chemical modification of myelopectase and lactoperoxidase with 2-(O-methoxypolyethylene glycol)-4,6-dichloro-s-triazine (Activated PEG). Cell Biochem Fun. 1998;16:123–128. doi:10.1002/(SICI)1099-0844(199806)16:2<123:AID-CBF777>3.0.CO;2-4
38. Falchi A, Taddei M. PEG-dichlorotriazine (PEG-DCT): a new soluble polymer-supported scavenger for alcohols, thiols, phosphines, and phosphine oxides. Org Lett. 2000;2(22):3429–3431. doi:10.1021/ol0002222
39. Dahrous KA, Abd-Elkader OH, Fouda MMG, AL-Othman ZA, El-Faham A. Eco-friendly method for silver nanoparticles immobilized decorated silica: synthesis & characterization and preliminary antibacterial activity. J Taiwan Inst Chem Eng. 2019;95:423–431. doi:10.1016/j.jtice.2018.07.020
40. Wayne PA. CLSI Standard M07: Methods for Dilution Antimicrobial Susceptibility Tests for Bacteria That Grow Aerobically:. 11th ed. Clinical and Laboratory Standards Institute; 2018.
41. Textor T, Fouda MMG, Maltibig B. Deposition of durable thin silver layers onto polyamides employing a heterogeneous Tollens’ reaction. Appl Surf Sci. 2010;256(8):2337–2342. doi:10.1016/j.apsusc.2009.10.063
42. El-Rafie MH, El-Naggary ME, Ramadan MA, Fouda MMG, Al-Deyab SS, Hebeish A. Environmental synthesis of silver nanoparticles using hydroxypropyl starch and their characterization. Carbohydr Polym. 2011;86(2):630–635. doi:10.1016/j.carbpol.2011.04.088
43. El-Aassar MR, Hafez EE, Fouda MMG, Al-Deyab SS. Synthesis, characterization, and antimicrobial activity of poly(acrylonitrile-co-methyl methacrylate) with silver nanoparticles. Appl Biochem Biotechnol. 2013;171(3):643–654. doi:10.1007/s12010-013-0395-8
44. Wang B, Zhuang X, Deng W, Cheng B. Microwave-assisted synthesis of silver nanoparticles in alkalic carboxymethyl chitosan solution. Engineering. 2010;2:387–390. doi:10.4236/eng.2010.25050
45. Sharma S, Thakur M, Deb M. Synthesis of silver nanoparticles using N1, N3-diphenylbenzamidine by microwave irradiation method. J Exp Nanosci. 2007;2:251–256. doi:10.1080/17458080701753744
46. Stemplecoskie KG, Scaino JC. Light emitting diode irradiation can control the morphology and optical properties of silver nanoparticles. J Am Chem Soc. 2010;132:1825–1827.
47. Gunanathan S, Kalishwaralal K, Vaidyanathan R, et al. Biosynthesis, purification and characterization of silver nanoparticles using Escherichia coli. Colloids Surf B. 2009;74:328–335. doi:10.1016/j.colsurfb.2009.07.048
48. Yagci Y, Sahin O, Ozturk T, Marchi S, Grassini S, Sangermano M. Synthesis of silver/epoxy nanocomposites by visible light sensitization using highly conjugated thiophene derivatives. React Funct Polym. 2011;71:857–862. doi:10.1016/j.reactfunctpolym.2011.05.012
49. Abul QF, Shafiq A, Khan HM, et al. Antibacterial effect of silver nanoparticles synthesized using murraya koeigii (L.) against multi-drug-resistant pathogens. Bioinorg Chem Appl. 2019;2019:4649506. doi:10.1152/japplgeny.1975.229.6.1641
50. Bhattacharya D, Samanta S, Mukherjee A, et al. Antibacterial activities of polyethylene glycol, tween 80 and sodium dodecyl sulphate coated silver nanoparticles in normal and multi-drug resistant bacteria. J Nanosci Nanotechnol. 2012;12:1–9. doi:10.1166/jnn.2012.6148
51. Wang AL, Yin HB, Ren M, Cheng Q, Zhou F, Zhang XF. Effect of different group containing organics on morphology controlled synthesis of nanoparticles at room temperature. Acta Metallurgica Sinica (Engl Lett). 2008;19(5):362–370. doi:10.1016/S1006-7719(06)62074-7
52. Popa M, Pradell T, Crespo D, Calderón-Moreno JM. Stable silver colloidal dispersions using short chain polyethylene glycol. Colloids Surf A. 2007;303(3):184–190. doi:10.1016/j.colsurfa.2007.03.050