Maximum fluorescence and electron transport kinetics determined by light-induced fluorescence transients (LIFT) for photosynthesis phenotyping

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Abstract
Photosynthetic phenotyping requires quick characterization of dynamic traits when measuring large plant numbers in a fluctuating environment. Here, we evaluated the light-induced fluorescence transient (LIFT) method for its capacity to yield rapidly fluorometric parameters from 0.6 m distance. The close approximation of LIFT to conventional chlorophyll fluorescence (ChlF) parameters is shown under controlled conditions in spinach leaves and isolated thylakoids when electron transport was impaired by anoxic conditions or chemical inhibitors. The ChlF rise from minimum fluorescence ($F_o$) to maximum fluorescence induced by fast repetition rate ($F_{m-FRR}$) flashes was dominated by reduction of the primary electron acceptor in photosystem II ($QA^-$. The subsequent reoxidation of $QA^-$ was quantified using the relaxation of ChlF in 0.65 ms ($F_{r1}$) and 120 ms ($F_{r2}$) phases. Reoxidation efficiency of $QA^-$ ($F_{r1}/F_v$, where $F_v = F_{m-FRR} - F_o$) decreased when electron transport was impaired, while quantum efficiency of photosystem II ($F_v/F_m$) showed often no significant effect. ChlF relaxations of the LIFT were similar to an independent other method. Under increasing light intensities, $F_{r2}'/F_q'$ (where $F_{r2}'$ and $F_q'$ represent $F_{r2}$ and $F_q$ in the light-adapted state, respectively) was hardly affected, whereas the operating efficiency of photosystem II ($F_q'/F_m'$) decreased due to non-photochemical quenching. $F_{m-FRR}$ was significantly lower than the ChlF maximum induced by multiple turnover ($F_{m-MT}$) flashes. However, the resulting $F_v/F_m$ and $F_q'/F_m'$ from both flashes were highly correlated. The LIFT method complements $F_v/F_m$ with information about efficiency of electron transport. Measurements in situ and from a distance facilitate application in high-throughput and automated phenotyping.

Keywords Fluorescence transient · Photosynthesis · Fast repetition rate · Electron transport kinetics

Introduction
Photosynthetic processes, from light absorption by the chlorophyll-based pigments through charge separation in the photosystem II (PSII) reaction centers and sequential electron transport, are related to the redox state of the primary quinone electron acceptor ($QA$) and coupled to the signature of chlorophyll fluorescence (ChlF) (Kautsky and Hirsch 1931; Baker 1931; Müh et al. 2012). Based on ChlF, parameters such as the maximum quantum efficiency of PSII ($F_v/F_m$) and non-photochemical quenching (NPQ) estimating the proportion of absorbed light energy utilized for PSII photochemistry and non-photochemical energy dissipation, respectively, were established (Butler 1978; Baker 2008; Lazár 2013). The quick assessment of ChlF makes this signal a powerful tool for plant phenotyping (Furbank and Tester 2011; Fiorani and Schurr 2013). Phenotyping
requires characterization of a large plant set which needs to be completed before significant changes in the measured traits occur. This is particularly difficult when phenotyping photosynthesis because this process is highly dynamic and sensitive to environmental conditions (Ananyev et al. 2005a; Kono and Terashima 2014).

In order to determine $F_v/F_m$, the ChlF signal is compared under conditions when $Q_A$ is in a fully oxidized state resulting in minimal ChlF ($F_o$) respective to the fully reduced state resulting in maximal ChlF ($F_m$). Two approaches using strong light pulses are widely accepted to reduce $Q_A$ fully: the single turnover flash (STF) and the multiple turnover flash (MTF) (Kalaji et al. 2017). A saturating STF has to provide high enough excitation power to induce one single charge separation in all PSII reaction centers and fully reduce $Q_A$ in order to yield maximum ChlF level ($F_m$) (Malkin and Kok 1966; Schreiber 1986a; Samson and Bruce 1996; Kolber et al. 1998; Steffen et al. 2001). The excitation flash needs to be short enough (from fs to few µs) to prevent reoxidation of $Q_A^-$ and reexcitation of PSII reaction centers (Malkin and Kok 1966; Belyaeva et al. 2014). In contrast, a saturating MTF requires at least 0.2-s duration of excitation at a few 1000 µmol photons m$^{-2}$ s$^{-1}$ (Ögren and Baker 1985; Schreiber et al. 1986b; Schreiber 2004). Within this time range, $Q_A$ is reduced and reoxidized several times followed by reduction of the plastoquinone (PQ) pool and electron transfer to Photosystem I (PSI) (Vernotte et al. 1979; Schansker et al. 2005). MTFs ultimately result in an about 50% higher maximum ChlF level ($F_{m-MT}$) compared to $F_{m-ST}$ (Schreiber 1986a; Schreiber et al. 1986b; Schansker et al. 2011). The difference between $F_{m-ST}$ and $F_{m-MT}$ was named the thermal phase because it is dependent on temperature, i.e., it is rate-limited (Delosme 1967). Later, the ChlF rise of the thermal phase was related to electron transport kinetics, particularly the accumulation of secondary quinone acceptors ($Q_B$) in a reduced state (Strasser et al. 1995; Lazár 2006). However, the origin of the thermal phase is not yet localized due to the complex and overlying kinetics of different electron transport processes (Rascher and Nedbal 2006; Mühl et al. 2012) and the probable involvement of additional ChlF quenchers (Schansker et al. 2011, 2014; Prášil et al. 2018; Magyar et al. 2018).

Alternatively, electron transport kinetics in the dark-adapted state were studied by following reoxidation of $Q_A^-$ coupled to ChlF relaxation after a STF (Vass et al. 1999; Petrouleas and Crofts 2005). According to an exponential decay model with three time constants, ChlF relaxes due to electron transport from $Q_A^-$ to $Q_B$ with a time constant ($\tau_1$) of 0.1–0.2 ms when the $Q_B$ site is occupied by a PQ (Bowes and Crofts 1980; Vass et al. 1999; Shinkarev 2004; Petrouleas and Crofts 2005). The second exponential component represents the reoxidation of $Q_A^-$ which had initially no PQ molecule bound (Taoka and Crofts 1990; Petrouleas and Crofts 2005). Therefore, this time constant ($\tau_2$) represents the binding of PQ molecule to the $Q_B$ site of PSII and is estimated to be between 2.2 and 10 ms (Vass et al. 1999; Eshaghi et al. 2000). The third component ($\tau_3$) is slow (500 ms to seconds) and interpreted as a back reaction from $Q_A^+$ to the donor side components of PSII, specifically the $S_2$ state of the oxygen evolving complex (OEC) (Robinson and Crofts 1983; Vass et al. 1999). Most of the existing ChlF-based techniques apply STFs either from a measuring head in direct contact with the leaf surface, or from a few cm distance. $F_{m-ST}$ is then recorded from a dark-adapted sample with an oxidized electron transport chain (Vernotte et al. 1979; Schansker et al. 2014). This allows standardized examination and modeling of ChlF relaxation kinetics in the dark (Vass et al. 1999). However, these requirements are impractical under conditions of ambient illumination, specifically when the presence of light is required for manifestation of stress conditions in the targeted plants. One such example is temperature stress, where low temperature enhances the photodamage effects of excess light (Pieruschka et al. 2010). In addition, the conventional fluorometric techniques may not provide sufficient resolution and throughput to capture highly dynamic regulation of photosynthesis.

The light-induced fluorescence transient (LIFT) method probes PSII from a distance using subsaturating (actinic) measuring flashlets in fast repetition rate (FRR) (Kolber et al. 1998; Osmond et al. 2017). In contrast to other techniques, no separate saturating flash is required in the LIFT method because the FRR probe flashlets are used directly for that purpose. The short measuring time of 0.2 s allows integration into automated systems for phenotyping in high spatio-temporal resolution. Following application in marine research (Kolber et al. 1998; Sijggett et al. 2001; Oxborough et al. 2012; Robinson et al. 2014), a stationary LIFT system was installed for monitoring plant canopy from a distance of 50 m using laser excitation (Pieruschka et al. 2010, 2014; Raesch et al. 2014). Operating efficiency of PSII ($F_{q'/F_m}$) measured with this previous LIFT system correlated well with pulse amplitude modulation (PAM) measurements ($R^2 = 0.89$) and CO$_2$ assimilation rates ($R^2 = 0.94$) (Ananyev et al. 2005a; Pieruschka et al. 2010, 2014).

Here, we evaluated a newly developed LIFT device for its capacity to yield robust fluorometric parameters useful in plant phenotyping. Parameters as maximum ChlF induced by FRR flash ($F_{m-FRR}$) and $Q_A^-$ reoxidation efficiency in 0.65 ms ($F_{q'/F_m}$) and 120 ms ($F_{q2}/F_m$) relaxation phases were introduced. The parameters were determined in isolated thylakoids and intact plants subjected to different treatments (electron transport inhibitors, anaerobiosis, or light) and approximated well-established ChlF parameters.

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Materials and methods

Plant cultivation

In total, 36 spinach (*Spinacia oleracea*) plants of genotype *Matador* were grown in the greenhouse in Jülich, Germany, under 16-h/8-h day/night cycle at 20 °C/18 °C. Light intensity was kept automatically between 60 and 300 µmol photons m⁻² s⁻¹ using additional lamps or shading nets. 400-mL pots were filled with a turf-clay substrate (ED73, Einheitserdewerke, Sinntal-Altengronau, Germany). Plants were watered automatically twice a day during cultivation. Measurements were performed using plants 28 or 32 days after sowing.

Isolation of thylakoids and PSII enriched membrane particles

For isolation of spinach thylakoids and PSII-enriched thylakoid membrane particles (BBY particles), fresh spinach leaves were bought from a local supermarket in Szeged, Hungary, and prepared as described in Berthold et al. (1981). For measurements with LIFT and FL3000, the final concentration of thylakoids was adjusted to equivalent chlorophyll *a* concentration of 10 µM (~ 10 µg mL⁻¹).

DCMU and DBMIB treatment

To manipulate the ChlF relaxation kinetics in thylakoid samples, we used 3-(3,4-dichlorophenyl)-1,1-dimethylurea (DCMU) and 2,5-dibromo-5-methyl-6-isopropyl-benzoquinone (DBMIB), which inhibit selectively the reoxidation of QA⁻ in PSII and of PQH₂ at the cytb₆f complex, respectively (Lazár et al. 2001; Kurisu et al. 2003). A thylakoid suspension of 3 mL was transferred to transparent plastic cuvettes. After 5-min dark-adaptation, DCMU and DBMIB were added to final concentration of 5 µM (1.17 µg mL⁻¹) and 0.66 µM (0.213 µg mL⁻¹), respectively. Samples were stirred manually and followed by either LIFT or FL3000 measurements using FRR flash for 0.75 ms (FRRF₀.₇₅₅₅ms) or STF, respectively. The number of technical replicates was 3–5. In addition, intact leaves were treated with DCMU to observe ChlF induction curves under conditions of blocked electron transport between QA and QB. Plants were dark-adapted overnight, then fully expanded leaves were left untreated or were subjected to 200 µM DCMU in 50 mL Milli-Q water (Tóth et al. 2005). The control was left untreated because a control with 1% ethanol in distilled water showed no effect on *F*ₘ and little effect on the ChlF rise compared to untreated leaves (Tóth et al. 2005). However, no ethanol was used in the DCMU solution to avoid possible side effects (Haldimann and Tsimilli-Michael 2005). DCMU was grinded to powder in order to dissolve it better in water. In the dark, one leaf per plant was left for 6 h in DCMU solution, then wiped and left for 30 min in the air. Measurements on attached leaves were done using 5 FRRFs₀.₇₅₅₅ms followed by one MTF for 750 ms (MTF₇₅₅₅ms). Only the first of the 5 FRRFs₀.₇₅₅₅ms is shown in the result section. Measurements were replicated with six different plants.

Anaerobic treatment under nitrogen atmosphere

Oxygen depletion inhibits the plastid terminal oxidase (PTOX), which normally keeps PQ in an oxidized state in the dark (Bohme et al. 1971; Cournac et al. 2000; Feilke et al. 2014). Anoxic treatment was used to manipulate the level of PQ reduction non-invasively in living plants (Tóth et al. 2007b). The anoxic atmosphere was maintained in the LI-COR 6400 transparent 2 × 3 cm chamber head (LI-COR, Inc., Nebraska USA) using nitrogen gas (N₂). Air inflow into the chamber came either from the ambient air (as control, with 400 ppm CO₂) or from N₂ gas supply without oxygen (containing < 1.5 ppm CO₂). The air flow rate during the measurements was 300 µmol air s⁻¹, and the block temperature of the LI-COR was kept at 20 °C. Prior to measurements, plants were dark-adapted overnight. A fully expanded leaf was inserted into the chamber and measured with the LIFT instrument through the transparent front window. Measurements were started after 5-min exposure to control or N₂ atmosphere using 5 FRRFs₀.₇₅₅₅ms followed by one MTF₇₅₅₅ms. After another 5 min, measurements were repeated using 5 FRR flashes for 2.5 ms (FRRFs₂.₅₅₅₅ms). Each first flash of the 5 FRRFs₀.₇₅₅₅ms and 5 FRRFs₂.₅₅₅₅ms is shown the result section. Measurements were replicated with six different plants.

Light response curves

To study electron transport kinetics of light-adapted plants, control plants of the N₂ atmosphere experiment were subjected to increasing levels of blue light provided by the LED (445 nm) light source of the LIFT instrument. The size of the illumination spot was around 3 cm². A light response measurement consisted of a total of 160 FRRFs₀.₇₅₅₅ms triggered at a 5-s interval. At every light intensity (30, 100, 300, 700 µmol photons m⁻² s⁻¹), ChlF was monitored over a period of 200 s by applying 40 consecutive FRRFs₀.₇₅₅₅ms. Light response curves were replicated with six different plants.

Fluorescence measurements

Different methods have been developed to separate absolute ChlF intensity and background radiation from relative
changes of ChlF yield due to Q_A reduction. This allows comparison of minimal ChlF (F_o in dark-adapted state, and F' in background light) and maximal ChlF (F_m in dark-adapted state, and F'' in background light) at initial redox state of Q_A and when Q_A is fully reduced at the end of a saturating flash, respectively (Schreiber et al. 1986a; Strasser et al. 1995; Kolber et al. 1998). Two different fluorometers as described below were used in this study. The LIFT method requires no additional saturating light pulse besides the measuring flashlets. Therefore, it is referred to as a modulated method. In contrast, the FL3000 fluorometer uses weak measuring pulses and a strong excitation flash. Therefore, it is referred to as a double-modulated method.

**FL3000 measurements**

ChlF relaxation after a STF was monitored by weak, non-actinic measuring flashes in increasing time intervals (Trtilek et al. 1997; Vass et al. 1999). These double-modulated ChlF measurements were performed with a FL3000 fluorometer (Photon Systems Instruments Ltd., Brno, Czech Republic) (Trtilek et al. 1997). The instrument is equipped with red LEDs (639 nm) for both actinic (20 µs with an excitation power of 1020 µmol photons m⁻² s⁻¹) and weak, non-actinic measuring flashes of 8 µs, with a measuring delay of 7 µs. Changes in ChlF yield can be measured in a very broad time range, from 100 µs to 100 s. Within this time range, reoxidation of QA usually shows three relaxation phases with corresponding time constants.

**LIFT measurements**

The newly developed compact LIFT instrument (Version LIFT-REM, Soliense Inc., New York, USA) is equipped with a blue LED (445 nm) excitation source. Excitation protocols composed of up to 7500 flashlets are used to manipulate the level of photosynthetic activity and ChlF (Fig. 1). ChlF emission is detected at 685 (± 10) nm. The LIFT device monitors any background signal in the detector range during inter-flashlet periods and subtracts this signal from the in-flashlet ChlF signal. The ChlF yield is internally normalized against excitation power of each flashlet to correct for smaller fluctuations. Flashlet excitation power along the entire FRR excitation phase is kept at a constant level. This was verified by observing a flat fluorescence transient using a fluorescence standard with constant quantum yield of fluorescence.

All measurements were done from a 0.6 m distance with flashlet length of 1.6 µs. The three used FRR flashes differ in the length of flashlet interval and ChlF induction phase (Table 1). FRRF₀.75ms consists of 300 flashlets with a 2.5-µs interval summing up to the 0.75 ms induction phase. The ChlF relaxation phase after the FRRF₀.75ms consists of 127 flashlets. The interval between those flashlets increases exponentially with

\[ j_i = 10^{1.28 + 0.0215x_i} \mu s, \]

where \( j_i \) is the interval length of the \( i \)th flashlet. For the DCMU and DBMIB experiments, the excitation power was approximately 20,000 µmol photons m⁻² s⁻¹ for the ChlF induction phase using FRRF₀.75ms. Due to the restricted excitation power of the LEDs, the FRRF₀.75ms lasts longer than a proper STF in order to get \( F_{m-FRR} \) saturated. For intact plants, the excitation power was 40,000, 24,000, and 1000 µmol photons m⁻² s⁻¹ for the ChlF induction phase using FRRF₀.75ms, FRRF₂.5ms, and MTF750ms, respectively. The interval between the flashlets for MTF750ms was extended from 2.5 to 100 µs due to exhausting of LED at
longer flashes. This resulted in the lower excitation power. Excitation power was measured at 1% duty cycle using a 5-s calibration flash measured by a quantum sensor (LI-190R, LI-COR, Inc.) and then extrapolated to 100%. For application of constant actinic light, the intensity of the blue LIFT LED in DC mode was calibrated using a quantum sensor (LI-190R, LI-COR, Inc.) at 0.6-m distance.

Analysis of LIFT and FL3000 raw data

For the calculation of $F_v/F_m$, the variable ChlF ($F_v$) is the difference between $F_m$ and $F_o$. In the LIFT analysis, $F_m$ is represented by $F_m-FRR$ as the averaged ChlF yield of 301st–302nd flashlet. $F_o$ is the ChlF yield of the first flashlet (Fig. 1). In the FL3000 analysis, $F_m$ is represented by $F_m-ST$ and measured 0.15 ms after the STF. $F_o$ is measured before the STF. The $Q_A^-$ reoxidation efficiency is calculated from the ChlF relaxation kinetics as follows:

$$F_{r1,2} F_v = \left( F_v \times t_{1,2} - \sum F_i \times j_i \right) \div \left( F_v \times t_{1,2} \right)$$

Here, $F_i$ is the ChlF yield in the relaxation phase at flashlet $i$. $F_i$ is multiplied by $j_i$ and summed up to represent the area of ChlF relaxation up to $t_1 = 0.65$ ms (for $F_{r1}$) and $t_2 = 120$ ms (for $F_{r2}$). In case of the FL3000 data, the time points for $t_1$ and $t_2$ were 0.52 ms and 100 ms after the STF, respectively, due to different relaxation protocols. The light-adapted states of $F_o$, $F_{m-FRR}$, $F_v$, and $F_{r1,2}$ are denoted as $F'$, $F_{m-FRR}'$, $F_v'$, and $F_{r1,2}'$, respectively.

Statistics

Analysis of variance (ANOVA) was used to calculate significant differences ($p < 0.05$) followed by Tukey’s test for pairwise comparison. Due to the small sample size ($n < 7$), normal distribution and homogeneity of variance were assumed. Analysis was done by R program using the multcomp package.

Results

Photosynthetic characteristics were studied by measuring light-induced ChlF transients using both the modulated LIFT and the double-modulated FL3000 device with an emphasis on the properties of electron transport from PSII towards PSI.

Determination of $F_m$ levels and electron transport kinetics

We used the FRRF₀.₇₅₅ₘₛ and MTF₇₅₀ₘₛ protocol to study the Induction of $F_m$–FRR and $F_m$–MT, respectively (Table 1). The $F_m$–FRR in control leaves was significantly lower than $F_m$–MT (Figs. 1, 2a, b). Under control conditions, $F_m$–FRR saturated earliest at about 0.25 ms depending on the excitation power (Fig. S1). At the end of the induction phase, a small peak of ChlF occurs pointing towards a minor ChlF quenching during the high excitation power of the FRRF₀.₇₅₅ₘₛ. These spikes contributed to $F_m$–FRR resulting in $F_v/F_m$ values independent of the excitation power. In contrast, $F_m$–MT in intact leaves was reached after 750 ms and showed the same $F_m$–MT as in the presence of DCMU (Fig. 2b).

We studied ChlF relaxation on attached leaves in the dark by using FRRF₀.₇₅₅ₘₛ under control and N₂ atmosphere. The absence of O₂ in the latter condition prevents reoxidation of the PQ pool in the dark (Tóth et al. 2007b). After 5 min in the N₂ atmosphere, ChlF relaxation phase was significantly altered compared to control conditions (Fig. 2a). While $F_v/F_m$ did not change under control and N₂ atmosphere, $F_v/F_m$ in the N₂ atmosphere decreased significantly to 0.16 (± 0.01) compared to 0.18 (± 0.01) in the control (Fig. 3). In contrast, $F_v/F_m$ was significantly increased in the N₂ atmosphere compared to the control. Notably, also the ChlF induction phase of MTF₇₅₀ₘₛ differed in N₂ atmosphere compared to control (Fig. 2b). Following MTF₇₅₀ₘₛ, we kept the plants for additional 5 min under the control or N₂ atmosphere. This allowed S-states of the OEC to relax in the dark (Kolber et al. 1998), whereas the PQ pool remained reduced in the N₂ atmosphere. Then a subsequent measurement using FRRF₂₅₃ₘₛ was initiated. $F_m$–FRR in the control

Table 1 Different excitation protocols are shown: Fast repetition rate flash for 0.75 ms (FRRF₀.₇₅₅ₘₛ) and for 2.5 ms (FRRF₂₅₃ₘₛ) as well as saturating multiple turnover flash for 750 ms (MTF₇₅₀ₘₛ). In the induction phase, flashlet interval is constant for given amount of flashlets and interval, while it increases exponentially in the relaxation phase to allow reoxidation of the primary quinone acceptor ($Q_A$). Flashlet length is always 1.6 µs and has in all flashes the same specified excitation power.

| Excitation protocol | Induction phase (ms) | Number of flashlets in induction phase | Flashlet interval in induction phase (µs) | Relaxation phase (ms) | Number of flashlets in relaxation phase | Flashlet length (µs) |
|---------------------|---------------------|---------------------------------------|------------------------------------------|----------------------|----------------------------------------|---------------------|
| FRRF₀.₇₅₅ₘₛ        | 0.75                | 300                                   | 2.5                                      | 209                  | 127                                    | 1.6                 |
| FRRF₂₅₃ₘₛ          | 2.5                 | 1000                                  | 2.5                                      | 209                  | 127                                    | 1.6                 |
| MTF₇₅₀ₘₛ           | 750                 | 7500                                  | 100                                      | 1975                 | 127                                    | 1.6                 |
treatment was reached at around 750 µs and then the ChlF signal started to decline (Fig. 2c). In the N₂ atmosphere, the ChlF level continued to increase during the FRRF₂.₅ms without reaching saturation. This resulted in a significantly increased $F_{m\rightarrow FRR}$ relative to the control. In addition, $F_o$ levels were higher in the N₂ atmosphere compared to control in the subsequent flashes (Fig. 2b, c). This led to significantly lowered $F_v/F_m$ under N₂ atmosphere using FRRF₂.₅ms ($p$ value < 0.001). In summary, $F_{m\rightarrow MT}$ induction in untreated leaves using MTF₇₅₀ms was confirmed by the same $F_m$ induced in DCMU-treated leaves. Full saturation of $F_{m\rightarrow FRR}$ level was confirmed using FRRF₂.₅ms under two conditions: (1) plants were in controlled, aerobic conditions (PQ pool was oxidized); and (2) the leaf was fully dark-adapted (OEC mainly in the $S_1$-state) (Delosme and Joliot 2002). The influence of increased PQ pool reduction was reflected in decreased $F_{r1}/F_v$ and increased $F_o$ and $F_{m\rightarrow FRR}$.

**Comparison of electron transport kinetics measured by the LIFT and FL3000 device**

We compared the ChlF relaxation kinetics acquired by the modulated LIFT to those of the double-modulated FL3000 device. For that purpose, we treated the thylakoids with different electron transport inhibitors. The two methods resulted in similar ChlF relaxation curves (Fig. 4). The $F_v/F_m$ values calculated from the LIFT measurements ranged between 0.58 ($±$ 0.01) for BBY and 0.7 ($±$ 0.02) for thylakoids (Fig. 5a). The FL3000 device showed generally lower $F_v/F_m$ values: 0.31 ($±$ 0.04) for BBY particles and 0.49 ($±$ 0.02) for thylakoids (Fig. 5b). In BBY particles, electron transport is impaired after the $Q_B$ site since the PQ pool is partly, and the PSI fully removed (Berthold et al. 1981). DBMIB binds to the cytb₆f complex, which blocks the reoxidation of the PQ pool (Bohme et al. 1971). Consequently, BBYs and DBMIB-treated thylakoids showed slower ChlF relaxation kinetics compared to thylakoids (Fig. 5c). The FL3000 device showed generally lower $F_{r1}/F_v$ and $F_{r2}/F_v$ values in both methods (Fig. 5c–f), $F_{r1}/F_v$ for the LIFT device ranged from 0.21 ($±$ 0.008) for thylakoids to 0.04 ($±$ 0.008) for BBY (Fig. 5d). The $F_{r1}/F_v$ values calculated from the FL3000 measurements were in general higher (e.g., 0.34 ($±$ 0.025) for thylakoids and 0.1 ($±$ 0.013) for BBY, Fig. 5d) but showed the same tendency as in the LIFT measurements. $F_{r2}/F_v$ showed increasing difference between the control and treated samples reflecting impaired electron transport (Fig. 5e, f). In summary, $F_{r1}/F_v$ and $F_{r2}/F_v$ measured with both devices responded specifically to the treatments which block electron transport at different steps.
Electron transport kinetics measured under ambient light

We measured light-response curves on attached spinach leaves in order to follow the light saturation of electron transport rate. $F'$ initially increased with increasing light intensities, but this increase was then reversed, most likely due to NPQ formation (Fig. 6a). Simultaneously, $F_{m'} - F_{R''}$ decreased in response to increasing light intensities due to NPQ formation. $F_{m'} - F_{R''}$ showed smaller absolute differences than $F_{m'} - F_{MT'}$, compared to the corresponding dark-adapted states (Fig. 6b).

The $F_{q'/F_{m'}}$ values we obtained with the FRRF$_{0.75ms}$ protocol correlated highly with $F_{q'/F_{m'}}$ values we retrieved with the MTF$_{750ms}$ protocol ($r^2 = 0.99$) during the measurements of blue-light response curves (Fig. 7). This demonstrates that FRRF$_{0.75ms}$ and MTF$_{750ms}$ measurements result in basically the same parameters with the exception of an offset, at least under these standard conditions. Increasing light intensities resulted in a significant decrease in the $F_{q'/F_{m'}}$ (Fig. 8a). In contrast, $F_{r1'/F_{q'}}$ and $F_{r2'/F_{q'}}$ were less affected by the higher light intensities (Fig. 8b, c). Upon dark to light transition at 30 µmol photons m$^{-2}$ s$^{-1}$, $F_{r1'/F_{q'}}$ decreased from 0.18 (± 0.009) to 0.10 (± 0.023) and increased to 0.2 (± 0.01) at the last two light intensities. $F_{r2'/F_{q'}}$ increased from 0.81 (± 0.018) to 0.95 (± 0.035) and then was stabilized at 0.89 (± 0.02) at the higher light intensities. $F_{q'/F_{m'}}$ measured in the light was not significantly different from dark-adapted values, with the exception of the initial light step at 30 µmol photons m$^{-2}$ s$^{-1}$. In summary, ChlF relaxation kinetics in the light were little affected by increasing light intensities and NPQ, whereas $F_{q'/F_{m'}}$ decreased.

Discussion

We induced ChlF transients by using different LIFT-FRR excitation protocols (FRRF$_{0.75ms}$ and MTF$_{750ms}$) at 0.6 m distance (Fig. 1). $F_{r}/F_{m}$ and $F_{q'/F_{m'}}$ were highly correlated between the two protocols (Fig. 7). Furthermore, we characterized a range of photo-physiological properties with an emphasis on the kinetics of electron transport from PSII towards PSI. These kinetics are determined by the well-established architecture of photosynthetic linear electron transport chain and can be observed via ChlF relaxation.
reflecting the kinetics of QA\(^{-}\) reoxidation (Vass et al. 1999). Efficiency of QA\(^{-}\) reoxidation was assessed in 0.65 ms (\(F_{r1}/F_v\), c, d), and 120 ms (\(F_{r2}/F_v\), e, f) relaxation phases in dark-adapted state acquired by light-induced fluorescence transient (LIFT) and double-modulated FL3000 fluorometer. Measurements were carried out on isolated spinach thylakoids and BBY particles. Thylakoid samples (10 µg chlorophyll/mL) were either untreated or treated with 5 µM 3−(3,4-dichlorophenyl)-1,1-dimethylurea (DCMU) or 0.66 µM 2,5-dibromo-5-methyl-6-isopropyl-benzoquinone (DBMIB), resulting in different chlorophyll fluorescence relaxations as shown in Fig. 4. Black diamonds show mean values and error bars indicate the 95% confidence intervals. Individual data points are depicted by a grey point (\(n=5\), except DCMU FL3000 and DBMIB FL3000 \(n=3\)). Means labeled with different letters differ significantly from each other according to Tukey’s multiple comparisons of means reflecting the kinetics of QA\(^{-}\) reoxidation (Vass et al. 1999). Efficiency of QA\(^{-}\) reoxidation was assessed in 0.65 ms and 120 ms relaxation phases after the FRR excitation, expressed in the \(F_{r1}/F_v\) and \(F_{r2}/F_v\) parameter, respectively. These simple parameters reflect the overall reoxidation of QA\(^{-}\) during the indicated time periods. In the light, \(F_{r2}/F_v\) reflected...
electron transport capacity from QA towards PSI and was far less sensitive to increasing light intensities than $F'/F_m'$ (Fig. 8). The results provide additional information about electron transport, which are not reflected by the $F_v/F_m$ parameter.

**Maximum fluorescence**

We demonstrated ChlF induction at 0.6 m distance by using the LIFT instrument on attached spinach leaves. $F_{m-MT}$ in the control leaves was reached at 750 ms after multiple turnover of PSII reaction centers (Fig. 2b). That $F_{m-MT}$ was indeed saturated by using MTF$_{750ms}$, was confirmed by using the DCMU treatment, which showed the same ChlF level. DCMU inhibits QA$^{-}$ reoxidation and induces $F_{m-MT}$ in intact leaves (Tóth et al. 2005). In contrast to $F_{m-MT}$, $F_{m-ST}$ is reached within 40 to 60 µs within one full turnover of the PSII reaction centers (Kolber et al. 1998; Nedbal et al. 1999; Belyaeva et al. 2014). The level of $F_{m-ST}$ is about 50% lower compared to $F_{m-MT}$ and is based on fully reduced QA$^{-}$ (e.g., Samson and Bruce 1996; Schansker et al. 2014). The difference between $F_{m-MT}$ and $F_{m-ST}$ is suggested to be caused by additional ChlF quenchers that are removed during multiple turnovers (Delosme 1967; Kalaji et al. 2017). In this study, the $F_{m-FRR}$ induced by FRRF$_{0.75ms}$ saturated at about 0.25 ms at highest excitation power of 40,000 µmol photons m$^{-2}$ s$^{-1}$ (Fig. S1). Within this time range, the OEC is already in the second turnover and QA$^{-}$ is once reoxidized by QB (Ananyev and Dismukes 2005b; Pérez-Navarro et al. 2016). The reoxidation of QA$^{-}$ by QB and QB$^{-}$ has time constants of 0.2 ms and 0.7 ms, respectively (Bowes and Crofts 1980; de Wijn and van Gorkom 2001; Tomek et al. 2001). The second time constants may vary with respect to the kinetics of H$^+$ uptake by QB$^{-}$ (Petrouleas and Crofts 2005). The saturating behavior of $F_{m-FRR}$ strongly indicates that photochemical processes stabilized within the FRRF$_{0.75ms}$ (Fig. S1). In agreement, the derived $F/F_m$ values were independent from the used excitation power. We conclude that $F_{m-FRR}$ reflected fully reduced QA$^{-}$ mainly associated with QB$^{-}$.
Using the FRRF\textsubscript{2.5ms} protocol, \( F_{m-FRR} \) declined after reaching a plateau under control conditions (Fig. 2c). This behavior was only observed when the PQ pool was oxidized and the sample was dark-adapted. Similarly, the polyphasic ChlF rise during a MTF of 15,000 \( \mu \text{mol photons m}^{-2} \text{s}^{-1} \) records a local ChlF maximum at about 2 ms (J-step) (Schreiber 1986a; Toth et al. 2007a; Schansker et al. 2011). This ChlF peak appears only when the sample is in the \( S_1 \)-state, i.e., dark-adapted (Strasser 1997). In the ChlF decline, \( Q_A \) reduction is overcome by \( Q_A^- \) reoxidation via the oxidized PSII primary donor (P680\(^+\)), which accumulates during the slow \( S_1-S_4 \) transition of the OEC (Schansker et al. 2011; Kalaji et al. 2017). The phase of ChlF decline matches time wise with the formation of \( Q_B^- \) (Bowes and Crofts 1980; de Wijn and van Gorkom 2001). After 2 ms, electron delivery of the OEC proceeds and re-reduction of \( Q_A^- \), further accumulation of \( Q_B^- \) and exchange of \( Q_A^- \) by an oxidized PQ takes place (Petrouleas and Crofts 2005). These processes lead to an increasing ChlF signal during a MTF, known as thermal phase (Delosme 1967; Lazár 2006).

It was shown before that PQ pool reduction leads to a higher ChlF signal by releasing non-photochemically quenched ChlF (Vernotte et al. 1979; Haldimann and Tsimilli-Michail 2005). In agreement, \( F_{m-FRR} \) increased without reaching saturation when the PQ pool was already reduced in the dark (Fig. 2c). Similarly, \( F_o \) under \( N_2 \) atmosphere was also higher than in the control but was the same in the subsequent FRRF\textsubscript{2.5ms} (\( p \) value = 0.403, data not shown) due to reoxidation of PQ pool during the flash. \( F_o \) yield was shown to be dependent on PQ redox state and can be recovered by far-red light pulse (Diner 1977; Hohmann-Marriott et al. 2010; Kalaji et al. 2017). However, this additional ChlF quenching is probably not directly controlled by the PQ pool (Tóth et al. 2005). It may be induced by the occupancy of the \( Q_B \)-pocket or by conformational changes in the PSII complex (Falkowski et al. 2004; Schansker et al. 2014; Magyar et al. 2018; Prášil et al. 2018). This might explain that \( F_{m-ST} \) (when only \( Q_A \) is reduced) cannot surpass the ChlF level at the J-step (Schreiber 1986a). Another reason for the lower \( F_{m-ST} \) compared to the J-step might be that STF induced a quenching mechanism during the reduction phase of \( Q_A^- \). At the end of the FRFR\textsubscript{0.75ms} induction phase, when changing from fast to low repetition rate flashlets, we noticed an instantaneous ChlF spike (Fig. S1). This indicates a fast-relaxing quenching mechanism. It was suggested earlier that carotenoid triplets quench ChlF within \( \mu \text{s} \) when operating with flashes at high excitation power (Schödel et al. 1999; Steffen et al. 2001; Braslavsky and Holzwarth 2012; Belyaeva et al. 2014). In summary, saturated \( F_{m-FRR} \) differs from \( F_{m-ST} \) in the reduction of \( Q_A^- \) to \( Q_B^- \) while \( Q_A^- \) is fully re-reduced. \( F_{m-FRR} \) and \( F_{m-ST} \) are expected to be comparable since \( Q_A^- \) is not known to quench any ChlF (Schansker et al. 2011). The saturation of \( F_{m-FRR} \) after 0.25 ms indicates that \( Q_B^- \) was not formed. The \( F_{m-FRR} \) differs from the J-step in the redox state of the OEC and the accumulation of \( Q_B^- \) which influence ChlF (see also Osmond et al. 2017). When reaching \( F_{m-MTF} \), at least one additional quencher is removed increasing ChlF signal by a still unclear mechanism (Magyar et al. 2018; Prášil et al. 2018).

### Validation of electron transport kinetics

Anaerobiosis inhibits PQ pool reoxidation in the dark (Bohme et al. 1971; Cournac et al. 2000; Feilke et al. 2014). As expected, reduced PQ pool under \( N_2 \) atmosphere affected ChlF relaxation and decreased \( F_v/F_o \) compared to control (Fig. 3b). ChlF relaxation was compared between the modulated LIFT-FRR and the double-modulated FL3000 device. The two devices measured similar qualitative responses comparable to earlier studies (Deák et al. 2014). However, \( Q_A^- \) reoxidation was faster in the first milliseconds when measuring with the double-modulated FL3000 than the LIFT device (Fig. 4). This might be due to the shorter duration of the STF than FRFR\textsubscript{0.75ms} where the latter reduced \( Q_B^- \) already. In addition, the FRR flashlets have an actinic effect, which partially reduce \( Q_A^- \) and slow down the \( Q_A^- \) reoxidation kinetics. The STF of the FL3000 device with 1000 \( \mu \text{mol photons m}^{-2} \text{s}^{-1} \) was probably not saturating resulting in the lower \( F_v/F_m \) values compared to the values derived by the LIFT device. In summary, a wide range of impaired electron transport processes were detected using the ChlF relaxation parameters \( F_{v1}/F_o \) and \( F_{v2}/F_v \) derived either by the LIFT or the FL3000 device.

### Measurements in the light and effect on electron transport kinetics

In the dark, the \( F_{m-FRR} \) yield was linked to PQ redox state and \( S \)-state of the OEC. The situation is different in the light because NPQ occurs and \( S \)-states are randomized. \( F_{m-FRR}' \) was interpreted as equilibrium of \( Q_A^- \) reduction, NPQ and subsequent \( Q_A^- \) reoxidation (Fig. 6a, Osmond et al. 2017). Accordingly, \( F_{m-MTF}' \) in the light does not reach saturation because NPQ limits light harvesting and electron transport quenches ChlF very efficiently (Loriaux et al. 2013). The MTF\textsubscript{750ms} with rather low excitation power (1000 \( \mu \text{mol photons m}^{-2} \text{s}^{-1} \)) limited \( F_{m-MTF}' \) saturation additionally. However, \( F_{q'}/F_{m'} \) derived by FRFR\textsubscript{0.75ms} during the measurement of blue-light response curve correlated highly with \( F_{q'}/F_{m'} \) derived by MTF\textsubscript{750ms} \((R^2 = 0.99) \) (Fig. 7). Correspondingly, \( F_{q'}/F_{m'} \) derived with a previous LIFT device using FRR flashes was shown to be well-correlated to the values measured by a PAM device \((R^2 = 0.89) \) (Pierschka et al. 2010, 2014). Comparable results were also shown by Samson et al. (1999) who carried out a similar experiment using STF and MTF. Dark-light transition at 30 \( \mu \text{mol photons m}^{-2} \text{s}^{-1} \)
was clearly separated based on the transient shape and the derived \( F_{r1}/F_{m}' \) and \( F_{r2}/F_{q}' \) values (Fig. 8). The strong initial response upon illumination appears to represent transition from the dark-adapted state of inactive electron transport to a light-stimulated state of engaged electron transport, e.g., activation of RUBISCO in the first few minutes of light acclimation (Kono and Terashima 2014). In conclusion, \( F_{q}'/F_{m}' \) derived by FRRF\(_{0.75ms}\) approximate \( F_{r2}/F_{q}' \) values derived by STF or MTF.

The \( F_{m-\text{ST}} \) and \( F_{m-\text{MT}} \) signal, along with \( F_{r}/F_{m} \) and NPQ, are the most common photosynthetic parameters used in ChlF-assisted plant phenomenics (Furbank and Tester 2011). These properties are relatively easy to measure by using existing ChlF techniques, and they are extremely sensitive to a range of photo-physiological properties of plants. At the same time, the \( F_{m} \) responses are rather non-specific, requiring additional information to identify the affected photosynthetic mechanisms (Kalaji et al. 2014). In addition, their direct responses to irradiance levels require that these parameters are measured under well-defined light conditions (generally in the dark, with pre-defined periods of dark adaptation), limiting their applications as reporters of physiological conditions under highly variable, natural light conditions. However, the properties of the photosynthetic electron transport from \( Q_{A} \) towards PSI (expressed as \( F_{r2}/F_{q}' \)) remain well-constrained under different ambient light intensities (Fig. 8c). The possibility for automation and measurements in the light using the LIFT method will make it possible to monitor the dynamics of photosynthetic traits under natural conditions.

**Conclusion**

Simultaneous measurements of \( F_{r}/F_{m} \) (respective \( F_{q}'/F_{m}' \)) and the kinetics of electron transport between PSII towards PSI provided additional information about the photosynthetic apparatus detecting differences in a wide range of physiological conditions. Performing these measurements non-invasively with high time resolution under natural environmental conditions has the potential to improve the efficacy of the photosynthetic phenotyping, while contributing to the advancement of knowledge about photosynthesis and its regulation.

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**Compliance with ethical standards**

**Conflict of interest** The authors declare that they have no conflict of interest.

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