Effect of Stoichiometry on Shape Memory Properties of Ti-Ni-Hf-Cu-Nb Shape Memory Alloys Manufactured by Suspended Droplet Alloying

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Abstract: A novel Ti-Ni-Hf-Cu-Nb shape memory alloy has been developed by a new combinatorial alloy synthesis method, the Suspended Droplet Alloying. The influence of alloying elements on the transformation temperature, the microstructure and the shape memory effect of this alloy have also been studied. It was found that Cu has a greater negative influence on the transformation temperature of Ti-Ni-Hf-Cu X alloys (about −5 K/at.%) than on the Ti-Ni-Cu X alloys (−0.67 K/at.%). In addition, the negative effect intensifies with increasing Hf content. The transformation temperature rapidly decreases with increasing Nb composition in the Ti-Ni-Cu-Nb and Ti-Ni-Hf-Cu-Nb alloys, with the solid solution of Nb in the matrix being 1 at.%. A Ti-Ni-Cu-Hf-Nb alloy with high thermal cycle stability has been developed, where the alloying elements affect the transformation behaviour via altering the slipping energy and forming different types of precipitations.

Keywords: martensitic transformation; shape memory alloys; Ti-Ni-Hf-Cu-Nb alloys; microstructure

1. Introduction

Shape memory alloys (SMAs) have attracted much attention since they were first reported by the U.S. Naval Ordnance Lab in 1962 due to their unique properties including shape memory effect (SME) and superelasticity (SE) [1–3]. Their unique characteristic is associated with a diffusionless reversible martensitic transformation (MT) between two phases: the martensite phase and the austenite phase, by introducing shape and volumetric changes at applied temperature or load [2]. The peculiar characteristics of SMAs make them a very promising candidate for structural and functional materials in a wide range of applications including aerospace, automotive, medical implants/instruments and energy industries [1,4]. Ti-Ni-based alloys are low cost, high performance SMAs, among which Ti-Ni-Cu and Ti-Ni-Nb alloys have good thermal cycling stability, but the hysteresis of Ti-Ni-Nb alloys is much larger than Ti-Ni-Cu alloys [2,5–9]. Chluba et al. [10] reported a Ti34Ni34Cu12 thin film with fine Ti2Cu precipitates showed very good thermo-mechanical stability due to the fine epitaxy Ti2Cu precipitates reducing Nb is not only an effective element to increase the thermo-mechanical stability, but also increases the hysteresis of transformation (by 100 K with 10 at.% Nb) in Ti-Ni-based SMAs, due to fine β-Nb particles around the precipitates [6]. Another element, Hf, is effective for increasing the martensitic transformation start temperature (Ms) of Ti-Ni-based SMAs. However, its addition reduces the recoverable deformation and increases the hysteresis of transformation (up to 200 K
at 30 at.% Hf) [2], which is the difference between martensitic transformation finish temperature (M_f) and the austenitic transformation starts temperature (A_s). The combination of good thermal cycling stability, low hysteresis and high MT has been a great challenge for several decades, due to the complex microstructure and high chemical composition sensitivity of SMAs [11].

Recently, a few quaternary SMAs have been developed to improve shape memory performance. Vermaut et al. studied Ti_{25}Ni_{50−x}Zr_{25}Cu_x SMAs with high Cu content (15 and 20 at.%) produced by melt spinning, but the application is limited by the process and semi-crystallised microstructure [12]. Lin et al. reported a linear relationship between M_s transformation enthalpy during cooling (ΔH_c) and Cu content in Ti_{30}Ni_{25−x}Pd_{25−y}Cu_{x+y} high temperature SMAs [13], but the cost of this alloy is expensive. Meng et al. studied the thermal stability, transformation temperatures (TTs), and microstructure of Ti_{36}Ni_{49−x}Hf_{15}Cu_x alloys, but the addition of Cu shows a small improvement on its thermal stability [14–16]. Wang et al. studied the influence of Nb content on the Ti_{50}Ni_{50}Cu_{10} SMA [17]; Tong et al. successfully produced a Ti-Ni-Cu-Nb alloy with stable thermal cycles via arc-melting [18]. However, the TTs of Ti-Ni-Cu-Nb alloys have not been improved comparing to Ti-Ni alloys. Yi et al. studied the Ti-Ni-Hf-Ag alloys [19] and Ti-Ni-Hf-Y alloys [20] with high transformation temperature but the cost of these doping elements are expensive. Li et al. fabricated Ti-Ni-Hf-Cu thin-film specimens and studied the phase transformation behaviour, but the use of thin film materials for structural applications is challenging [21]. Yi et al. improved the mechanical properties of Ti-Ni-Hf-Cu via introducing (Ti, Hf)_2Ni precipitates, which indicates a potential way to improve the shape recovery of Ti-Ni based SMAs [22]. Lemke et al. [23] and Chen et al. [24] reported that the Ti-Ni-Hf-Nb alloys have good workability and competitive TTs, indicating the possibility of further development of Ti-Ni-based SMAs.

From the aforementioned research, it is seen that the quaternary or more complex SMAs system has not been sufficiently investigated. One of the reasons is the difficulty of producing a vast quantity of complicated alloys with precision in a short time frame. Hence, we developed a rapid alloying system, Suspended Droplet Alloying (SDA), to conquer the problem and systematically study the influence of alloying elements on the MT of Ti-Ni-based SMAs. The specifics of this technique are detailed in a previously published study [25]. Here, we proposed using Hf to improve the M_s while using Nb and Cu to adjust the hysteresis and thermal stability of Ti-Ni-based SMAs by prompting nucleation, forming various precipitates and altering the matrix composition. As such, Ti-rich Ti-Ni-Cu-Nb and Ti-Ni-Hf-Cu-Nb alloys were produced by the SDA process to develop high transformation temperature and high thermo-mechanical stability SMAs. This study focused on thermal cycle stability, MT behaviours and microstructure investigation of novel Ti-Ni-Hf-Cu-Nb SMAs developed via the SDA technique.

2. Materials and Experimental Methods

Elemental Ti, Ni, Cu, and Nb wires with 99.9% purity used in this study were provided by Advant Ltd. (High Wycombe, UK). The purity of Hf and Zr wires was 97% due to the difficulty of separating these elements from each other. Samples were fabricated using the Quantum SDA machine under Argon atmosphere (oxygen level 30 ppm) and then sectioned by Agie Charmilles CUT 20 P Electrical Discharge Machining (EDM), before being ground and polished (by 0.04 μm compound) for further investigation. JOEL 7000 and TESCAN field emission SEM, equipped with Oxford Inca Energy-Dispersive X-ray Spectroscopy (EDS) system was applied to investigate the microstructure and chemical composition with compositional accuracy around 1 wt.%. The micro-pillar test was performed by CFEI Quanta 3D FEG FIB-SEM with Alemnis nanoindentation system. The pillar was milled to 10 μm diameter, 32 μm height and the loading rate of compression test was 10 μN/s.

Samples were cut to a size of 1 mm ± 0.5 mm thickness and 4 mm ± 1 mm diameter for Differential Scanning Calorimetry (DSC) testing, after the EDS chemical composition
measurement. The heating and cooling rate of DSC testing were 10 K/min. The Mettler Toledo DSC1 system was applied to understand the transformation behaviour from 223 K to 473 K. The first thermal cycle of DSC testing was excluded from the results as it may have been affected by the sample preparation process. The TTs, including $M_s$, $M_f$, $A_s$ and austenite transformation finish temperature ($A_f$) were determined by the tangent method and the enthalpies were the area between the transformation peak and baseline in the DSC curve [26]. The typical error of transformation temperature measurement is about 10 K.

The Mettler Toledo DSC500 system was used to determine the heat treatment temperature by performing the test from RT (298 K) to 1773 K at 10 K/min. An isothermal period was held at 1773 K for 30 min before cooling at the same rate. The sample was protected by 99.995% purity Argon atmosphere during the experiment.

The samples were wrapped with Tantalum wire and sealed in a vacuumed steel tube to prevent oxidation during post heat treatment.

3. Results and Discussions

3.1. Influence of Hf and Cu Content on Transformation Temperature of Ti-Ni Based Shape Memory Alloys

$\text{Ti}_{51}\text{Ni}_{49}$, $\text{Ti}_{51}\text{Ni}_{30}\text{Cu}_{10}$, $\text{Ti}_{51}\text{Ni}_{24}\text{Cu}_{25}$, $\text{Ti}_{46}\text{Ni}_{40}\text{Hf}_5$ and $\text{Ti}_{36}\text{Ni}_{44}\text{Hf}_{15}\text{Cu}_5$ alloys were fabricated by SDA technique and studied by DSC testing after solution treatment, as shown in Figure 1. The addition of Hf reduced the thermal cycle stability (i.e., the $M_s$ decreases as the number of thermal cycles increases) of Ti-Ni-based SMAs, as the $M_s$ of $\text{Ti}_{51}\text{Ni}_{49}$ alloys decreased by 12 K after 8 cycles while the $M_s$ of $\text{Ti}_{46}\text{Ni}_{40}\text{Hf}_5$ alloys decreased by 18 K after the same number of cycles. Cu was shown to significantly improve the thermal stability of Ti-Ni alloys as both the $\text{Ti}_{51}\text{Ni}_{39}\text{Cu}_{10}$ and $\text{Ti}_{51}\text{Ni}_{24}\text{Cu}_{25}$ alloys displayed constant $M_s$ after 3 cycles. $\text{Ti}_{36}\text{Ni}_{44}\text{Hf}_{15}\text{Cu}_5$ SMA also had high $M_s$ and acceptable thermal stability, with the $M_s$ of this alloy decreasing by about 8 K in 8 cycles, surpassing the Ti-Ni and Ti-Ni-Hf SMAs. It is well known that the Cu is able to improve the thermal cycle stability of Ti-Ni-Cu SMAs by reducing the critical shearing stress ($c_{44}$) of the lattice [27]. The results in Figure 1 show that the Cu improves the thermal cycle stability of the Ti-Ni-Hf alloys with a small compromise of $M_s$.

![Figure 1](image-url) Reduction of $M_s$ of Ti-Ni-based SMAs with increasing thermal cycles. The labels Cu25, Cu10, Hf5, NiTi, Hf15Cu5 represent the thermal stability of $\text{Ti}_{51}\text{Ni}_{24}\text{Cu}_{25}$, $\text{Ti}_{51}\text{Ni}_{39}\text{Cu}_{10}$ $\text{Ti}_{46}\text{Ni}_{40}\text{Hf}_5$, $\text{Ti}_{51}\text{Ni}_{49}$ and $\text{Ti}_{36}\text{Ni}_{44}\text{Hf}_{15}\text{Cu}_5$, respectively.

Prior to creating quaternary and higher complicated alloys, it is necessary to investigate the influence of different elements (Cu, Hf, and Nb) on the $M_s$ and hysteresis. Figure 2 shows the influence of Hf composition on the $M_s$ of the solution treated (1273 K, 2 h
followed by water quench) Ti$_{51-x}$Ni$_{49}$Hfx SMA produced by the SDA process. The ($\Delta$M$_a$) in Figure 2 is the $M_s$ difference between Ti-Ni-Hf and Ti$_{51}$Ni$_{49}$ SDA samples at the same solution treated condition, while the Ref. $\Delta$M$_a$ is the difference between $M_s$ of Ti-Ni-Hf alloys reported in the literature and the $M_s$ of Ti$_{51}$Ni$_{49}$ SDA samples [2,9,28]. The effect of Hf content on $M_s$ can be simplified as a linear function, which is

$$M_s (\chi) = 16\chi + 333 \tag{1}$$

where $\chi$ is the atomic ratio of Hf content. Although a wide range of Ti-Ni-Hf SMAs have been reported, the manufacturing process and post treatment vary among them. Since the manufacturing process and post treatment has a strong impact on the shape memory behaviours, it is difficult to compare the shape memory behaviours between literature. In this study, we produced Ti$_{51}$Ni$_{49-\chi}$Hfx SMAs with Hf content from 0 to 30 at.% by the SDA technique and consistent heat treatment. It is found that, the Hf content has a linear influence on the $M_s$ and the coefficient is 16 K/at.%. Second, the $M_s$ of Ti-rich Ti-Ni-Hf alloys is slightly higher than the Ni-rich alloys. It is due to the Ti-rich alloys ensures the low Ni content in the matrix, which has a strong negative influence on the $M_s$ of Ti-Ni-based alloys [29]. It has been reported in the literature that the $M_s$ of Ni-rich Ti-Ni-Hf alloys increased after ageing, due to the forming of (Ti+Hf)$_4$Ni precipitates reduced the Ni content in the matrix [28]. In this study, the Ti-rich alloys show higher $M_s$ without complicated ageing treatment as they ensure low Ni content in the matrix. Third, Figure 2 shows the transformation hysteresis of Ti$_{51}$Ni$_{49-\chi}$Hfx SMAs increases with increasing Hf content. However, the hysteresis in SDA samples is around 100 K, which is about half of the hysteresis reported by literature [2,30]. It has been reported that the hysteresis in Ti-Ni-based SMAs is caused by the heterogeneity of the materials, including grain boundary and phase interface [31]. The microstructure of Ti-Ni-Hf samples shows that the grain size decreases with increasing Hf content, which could be the reason for the variant of hysteresis. Compared to the alloys produced by arc melting in Paul’s study [30], the samples fabricated by SDA process have smaller grain size and better homogeneity. The higher amount of Hf in the materials provided more nucleation sites and enlarged the temperature gradient during solidification, therefore the grain size was reduced.

![Figure 2](image-url)\n
**Figure 2.** Influence of Hf content on the transformation temperature of Ti$_{51}$Ni$_{49-\chi}$Hfx SDA built SMA at as-fabricated condition. The $M_f$, $A_s$ and $\Delta$M$_a$ are martensitic transformation finish temperature, austenite transformation start temperature and the difference between the $M_s$ of Ti$_{51}$Ni$_{49-\chi}$Hfx and the $M_s$ of Ti$_{51}$Ni$_{49}$ determined by the DSC experiment. The Ref. $\Delta$M$_a$ is the difference between the $M_s$ of Ti$_{50}$Ni$_{50-\chi}$Hfx and the $M_s$ of Ti$_{51}$Ni$_{49}$ reported by literature [2].
Previous studies detailed in literature and experimental results of SDA samples have shown that Hf significantly increases the $M_s$. Figure 3a shows the Cu decreases the $M_s$ marginally, with increasing composition, which is consistent with the results reported in the literature [16,32]. The influence of Cu content on the $M_s$ of solution treated (1073 K, 2 h followed by water quench) Ti$_{51}$Ni$_{49-x}$Cu$_x$ alloys can be described as:

$$M_s (\% ) = -0.67\% x + 344$$

(2)

where $x$ is Cu atomic composition and $M_s$ in K. It is well known that Cu is able to improve the thermal cycle stability and reduce the hysteresis of Ti-Ni-based SMAs, but the systemic study on its effect on the $M_s$ has not been reported. The results in Figure 3a and Equation (2) shows the Cu has a slightly negative effect on the $M_s$ of Ti-Ni-Cu alloys.

Figure 3a shows the Cu decreases the $M_s$ of Ti-Ni-Cu SMAs shows the $M_s$ decreases linearly with increasing Cu content, and (b) Ti$_{36}$Ni$_{49-x}$Hf$_{15}$Cu$_x$ (labelled as Ti-Ni-Hf15-Cux) and Ti$_{41}$Ni$_{49-x}$Hf$_{10}$Cu$_x$ (labelled as Ti-Ni-Hf10-Cux) alloys show linear decreasing $M_s$ as the Cu content increase in Ti-Ni-Hf-Cu alloys. The influence of Cu becomes stronger in higher Hf content alloys.

Figure 3b shows that Cu has a greater influence on the $M_s$ of quaternary Ti-Ni-Hf-Cu alloys than that of the ternary alloys. Although some researchers have investigated the transformation behaviour of Ti$_{36}$Ni$_{49-x}$Hf$_{15}$Cu$_x$ alloys, some compositions, including Ti$_{36}$Ni$_{49}$Hf$_{15}$Cu$_{10}$ and Ti$_{41}$Ni$_{49-x}$Hf$_{10}$Cu$_x$ have not been reported [33]. Therefore, Ti$_{36}$Ni$_{44}$Hf$_{15}$Cu$_5$, Ti$_{36}$Ni$_{39}$Hf$_{15}$Cu$_{10}$ and Ti$_{41}$Ni$_{49-x}$Hf$_{10}$Cu$_x$ alloys were fabricated. The influence of Cu content on $M_s$ of as-fabricated Ti-Ni-Hf-Cu alloys is shown in Figure 3b. The higher the Hf composition, the greater the reduction observed in of $M_s$ with increasing Cu composition. Meng et al. also found that Hf and Cu had a negative effect on each other [15]. For the Ti-Ni-Hf-Cu sample with 15 at.% Hf, the influence of Cu composition on $M_s$ can be described as:

$$M_s (\% ) = -5.7\% x + 445$$

(3)

where $x$ is the Cu content (atomic ratio) in Ti-Ni-Hf-Cu alloys with 15 at.% Hf. The Ti$_{36}$ Ni$_{44}$Hf$_{15}$Cu$_5$ thin-film sample reported in the literature was manufactured by the melt-spun process, which has different transformation characteristics to the bulk materials ($M_s$ is 420 K in as-cast condition, and 363 K after annealing) [14,34]. It shows good agreement between our study and the literature, the slightly higher $M_s$ in thin-film samples might result from a faster cooling rate and smaller grain size. The $M_s$ of 10 at.% Hf alloys also displayed linear reduction as the Cu content was increased and the SDA results agreed with the literature [33]. The relation can be described as:

$$M_s (\% ) = -5.1\% x + 380$$

(4)
where \( x \) is the Cu content (at.\%) in Ti-Ni-Hf-Cu alloys with 10 at.\% Hf. The results demonstrated that Hf and Cu have a negative influence on each other on TTS, as the \( M_s \) decreases in Ti\(_{41}\)Ni\(_{49-x}\)Hf\(_{10}\)Cu\(_x\) SMA (−5.7 K/at.\%) was much greater than seen with the Ti\(_{51}\)Ni\(_{49-x}\)Cu\(_x\) SMA (−0.67 K/at.\%). It is likely due to the Hf atom has a larger size, which increases the misfit of the matrix lattice and requires more energy for transformation, while the Cu atom in the matrix not only reduced the shearing resistance but also reduces the critical transformation energy. It has been reported that the lowest slip energy direction of Ti-Ni-Hf alloys is different from Ti-Ni and Ti-Ni-Cu alloys, which means the greater negative effect of Cu in Ti-Ni-Hf-Cu alloys might result from the Hf increases the atom distance in the lattice and reduces the slipping energy [35].

### 3.2. Improving the Thermal Stability of Quaternary and Quinary Shape Memory Alloys

It is known that the Nb composition not only affects the \( a_b \) but also the deformation strain of Ti-Ni-Nb alloys [6,7]. The \( a_e \) increases from 308 K to 364 K as the Nb composition increases from 3 at.\% to 20 at.\%, as shown in Figure 4a. The influence of Nb-content on the \( M_s \) can be described as:

\[
M_s (x) = -6.3x + 343
\]

where the \( x \) is atomic Nb content in Ti-Ni-Nb alloys. The influence of Nb composition on \( M_s \) of Ti\(_{51-0.5x}\)Ni\(_{49-0.5x}\)Nb\(_x\) alloys is slightly stronger than the Cu in Ti-Ni-Hf-Cu alloys (−6.3 K/at.\% for Nb and −5.7 K/at.\% for Cu). This is because the Nb solution in the matrix reduces the slipping energy. The atom size of Nb is larger than Cu, which means the influence of Nb content on the TTS is stronger.

![Figure 4. The influence of Nb content on the \( M_s \) of (a) Ti\(_{51-x}\)Ni\(_{51-x}\)Nb\(_{2x}\) alloys [7] and (b) Ti\(_{51-0.5x}\)Ni\(_{39-0.5x}\)Cu\(_{10}\)Nb\(_x\) alloys produced by SDA shows Nb decreases the \( M_s \).](image)

Ti-rich alloy based on Ti\(_{51}\)Ni\(_{39}\)Cu\(_{10}\) alloys were chosen to study the influence of Nb on the \( M_s \) of Ti-Ni-Cu-Nb alloys, as this composition has good thermal stability, narrow transformation peak, and small hysteresis [10]. The results in Figure 4b show the \( M_s \) of alloys decreased from 338 K to 283 K with increasing Nb composition from 1 at.% to 3 at.\%, while the thermal stability remained the same as the base alloy. This suggests that the increasing Nb composition decreases the \( M_s \), but does not negatively affect the thermal stability of Ti\(_{51-0.5x}\)Ni\(_{39-0.5x}\)Cu\(_{10}\)Nb\(_x\) SMA when the Nb content is lower than 3 at.\%. The influence of Nb content on the \( M_s \) can be described as:

\[
M_s (x) = -19x + 338
\]

where \( x \) is the Nb content (atomic ratio) in Ti-Ni-Cu-Nb alloys. Comparing the influence of Nb content on the Ti-Ni-Nb SMAs, a greater \( M_s \) reduction was seen in the Ti\(_{51-0.5x}\)Ni\(_{49-0.5x}\)Cu\(_{10}\)Nb\(_x\) alloys (−6.3 K/at.\% in Ti\(_{51-0.5x}\)Ni\(_{49-0.5x}\)Nb\(_x\) SMA). The negative effect on TTS in Ti-rich Ti-Ni-Cu-Nb SMA is greater than (Ti\(_{50}\)Ni\(_{40}\)Cu\(_{10}\))\(_{100-x}\)Nb\(_x\) SMA.
reported by Wang et al. [17]. Since our study shows Cu and Nb also have a negative effect on each other on the transformation temperature. The lower TTs in Ti-rich Ti-Ni-Cu-Nb alloys is the result of higher Nb and Cu content. As a consequence of these results, the addition of Nb-content will be limited in further alloy development.

The microstructure of $\text{Ti}_{51-0.5x}\text{Ni}_{39-0.5x}\text{Cu}_{10}\text{Nb}_x$ alloys reveal that the $\beta$-Nb particles present around the precipitates, and the EDS results show that the Nb content in the matrix does not exceed 1 at.% as the overall Nb content of the alloys increases as shown in Table 1. Figure 5 shows the backscatter images of Ti-Ni-Cu-Nb samples. The average spacing between $\text{Ti}_2(\text{Ni}, \text{Cu})$ precipitates (dark contrast particles) are approximately 20 $\mu$m and the area fraction of white particles (Nb-rich) expanded as the Nb content increased. Most Nb-rich particles appear around the $\text{Ti}_2(\text{Ni}, \text{Cu})$ precipitates. EDS analysis shows the typical Nb content around $\text{Ti}_2(\text{Ni}, \text{Cu})$ precipitates was 2 to 4 at.% (marked as 2 in Figure 5), while the Nb content in the matrix varied from 0.5 to 1.6 at.% as the Nb increased from 1 to 3 at.% (marked as 1 in Figure 5). No $\text{Ti}_2\text{Cu}$ precipitates were found in these samples and the Ni composition was higher than Cu in $\text{Ti}_2(\text{Ni}, \text{Cu})$ precipitates. The EDS point analysis on the $\beta$-Nb particles shows 6 to 26 at.% Nb content (marked as 3 in Figure 5). This indicates that the Nb affected the transformation behaviour mainly by changing the misfit between the matrix and precipitates, instead of altering the lattice parameters of the matrix like Cu and Hf do. The low magnification image Figure 5d shows the Ti-Ni-Cu-Nb samples built by SDA have a homogeneous microstructure and fine precipitates. Comparing to the ($\text{Ti}_{50}\text{Ni}_{40}\text{Cu}_{10})_{100-x}\text{Nb}_x$ SMA produced by arc melting, the maximum solution of Nb in matrix is similar (1.9 at.-% in arc melt sample) but the size of $\beta$-Nb particles in SDA samples are about 50% smaller [17]. This indicates better compositional homogeneity and finer grains in SDA Ti-rich samples.

Since Nb does not affect the thermal stability of Ti-Ni-Cu alloys and the reduction of $M_s$ is acceptable, the next step was to add Hf into the alloys to alter the transformation temperature and assess the thermal stability of Ti-Ni-Hf-Cu-Nb alloys. As shown in Figure 6, $\text{Ti}_{41-0.5x}\text{Ni}_{39-0.5x}\text{Cu}_{10}\text{Hf}_{10}\text{Nb}_x$ alloys were produced and investigated after 1273 K 3 h heat treatment. Eleven DSC cycles were applied to investigate the thermal stabilities of Ti-Ni-Hf-Cu-Nb samples. Figure 6a and c indicate that increasing Nb composition decreased the $M_s$ of these alloys but thermal stability remained at the same level. The $M_s$ decreased 25 K when the Nb content increase from 0 to 2 at.% as shown in Figure 6a. Figure 6c shows the $\text{Ti}_{41-0.5x}\text{Ni}_{39-0.5x}\text{Cu}_{10}\text{Hf}_{10}\text{Nb}_x$ alloys have good thermal stability as the difference of $M_s$ between each thermal cycle are $\sim$0.6 to $\sim$0.8 K/cycles, and the decrease of $M_s$ in 11 cycles was less than 10K. The $\text{Ti}_{41-0.5x}\text{Ni}_{39-0.5x}\text{Cu}_{10}\text{Hf}_{10}\text{Nb}_x$ alloys. Figure 6b shows the transformation heat of the third and eleven cycles of $\text{Ti}_{41-0.5x}\text{Ni}_{39-0.5x}\text{Cu}_{10}\text{Hf}_{10}\text{Nb}_x$ alloy, and Figure 6d shows the averaged heat difference between each thermal cycle. The transformation heat decreases with increasing thermal cycles and Nb content, which indicates less and less materials are transformable during thermal cycles. This usually results from the accumulation of dislocation and the pinning effect of precipitation. The hysteresis of these samples was lower than 40 K, similar to the Ti-Ni binary alloys (40 K). Therefore, the results indicate that Nb can be included in Ti-Ni-Cu and Ti-Ni-Hf-Cu alloys for improving thermal-mechanical fatigue properties.

| Sample | $\text{Ti}_{50}\text{Ni}_{39}\text{Cu}_{10}\text{Nb}_1$ | $\text{Ti}_{50}\text{Ni}_{38}\text{Cu}_{10}\text{Nb}_2$ | $\text{Ti}_{50}\text{Ni}_{37}\text{Cu}_{10}\text{Nb}_3$ |
|--------|-----------------------------------------------|-----------------------------------------------|-----------------------------------------------|
| 1 (matrix) | $\text{Ti}_{50}\text{Ni}_{42}\text{Cu}_8$ | $\text{Ti}_{50}\text{Ni}_{42}\text{Cu}_7\text{Nb}_1$ | $\text{Ti}_{50}\text{Ni}_{41.5}\text{Cu}_7.5\text{Nb}_1$ |
| 2 (Ti-rich) | $\text{Ti}_{55}\text{Ni}_{32}\text{Cu}_{11}\text{Nb}_2$ | $\text{Ti}_{60}\text{Ni}_{28}\text{Cu}_{6}\text{Nb}_2$ | $\text{Ti}_{60}\text{Ni}_{28}\text{Cu}_{6}\text{Nb}_3$ |
| 3 (Nb-rich) | $\text{Ti}_{50}\text{Ni}_{38}\text{Cu}_{11}\text{Nb}_2$ | $\text{Ti}_{55}\text{Ni}_{14}\text{Cu}_{5}\text{Nb}_{26}$ | $\text{Ti}_{60}\text{Ni}_{28}\text{Cu}_{6}\text{Nb}_4$ |
| Area analysis | $\text{Ti}_{50}\text{Ni}_{30}\text{Cu}_{10}\text{Nb}_1$ | $\text{Ti}_{40}\text{Ni}_{39}\text{Cu}_{10}\text{Nb}_2$ | $\text{Ti}_{50}\text{Ni}_{37}\text{Cu}_{10}\text{Nb}_3$ |
The dHs (Mₜₜ difference between cycles) increased with greater Nb composition up to 1 at.% but decreased as the Nb content reached 2 at.%, as shown in Figure 6c. Although the 1 at.% Nb alloy had the highest dHs, its thermal cycle stability was the poorest among these compositions, and the alloy with 2 at.% Nb addition was chosen for further research. The variation of transformation heat (enthalpies) increased as the Nb composition was raised, as shown in Figure 6b. Since the Nb composition in the matrix stayed constant when Nb content was higher than 1%, the transformation heat was significantly affected by Nb-rich particles around the precipitates. The alloys with 0.5 and 1 at.% Nb has the lowest enthalpies at the third thermal cycle while the alloys with 2 at.% Nb has the highest enthalpies. The slightly lower enthalpies in the 0.5 and 1 at.% alloys might be due to higher precipitate fraction in the materials. However, the sample with 2 at.% Nb has the largest enthalpies drop at the eleventh thermal cycle (about 3.5 J/g), which indicates the transformable materials in this alloy significantly reduced after 11 cycles. This could be the increasing size and quantity of the β-Nb particles hindered the recovery deformation. dH (ΔH), which is the transformation heat difference between austenite and martensitic transformation, is often used to analyse transformation behaviour. The dH of the third cycle in Ti₄₁−0.₅₉Ni₃₉−0.₅₉Cu₄Hf₁₀Nbₓ alloys linearly increased when the Nb composition was lower than 1 at.% but only rose slightly when the content was between 1 and 2 at.% as shown in Figure 6d. The dH of the eleventh cycle, however, linearly increased with
increasing Nb content. Both the transformation overall enthalpies and their difference reduced during thermal cycles in the low Nb content samples. The aforementioned results might be due to the reversible lattice deformation being suppressed by slips generated during thermal cycles, which introduced the residual martensite phase around the Nb-rich precipitates. This residual martensite phase reduces the nucleation energy of martensite and the reverse transformation. The variation of enthalpies shown in Figure 6d indicates the Nb content has a significant influence on the residual lattice deformation, as the transformation heat difference after 11 thermal cycles is larger in the sample with higher Nb content. The decreasing enthalpies and dH in low Nb samples is due to the irreversible lattice deformed and remaining deformation during the first few thermal cycles.

![Figure 6. Transformation behaviour of Ti41−0.5xNi39−0.5xCu10Hf10NbX SMA: (a) influence of Nb content and thermal cycles (the third cycle is indicated by the black line and the eleventh cycle is red line) on M_s temperature; (b) influence of Nb content and thermal cycles on transformation heat; (c) influence of Nb content on the reduction of M_s per thermal cycle; (d) influence of Nb content and thermal cycles on the transformation heat difference between austenite transformation and MT.](image)

3.3. Microstructure and Shape Memory Behaviours of Ti-Ni-Hf-Cu-Nb Alloys

3.3.1. Transformation Temperature of Ti-Ni-Hf-Cu-Nb SMAs

The DSC result in the Figure 7a,b reveal that two alloys with similar compositions, Ti_{43}Ni_{33}Hf_{10.5}Cu_{11}Nb_{2.5} and Ti_{44}Ni_{34}Hf_{10}Cu_{10}Nb_{2}, have different transformation behaviours under thermal cycles. Figure 7a shows the Ti_{43}Ni_{33}Hf_{10.5}Cu_{11}Nb_{2.5} alloy has a low transformation temperature (M_s = 248 K or −25 °C), but the transformation curves between thermal cycles are overlapped and the transformation peak is wide. Figure 7b shows the Ti_{44}Ni_{34}Hf_{10}Cu_{10}Nb_{2} alloy has a higher transformation temperature (M_s = 305 K or 32 °C), but the thermal stability of this alloy is weaker (M_s dropped from 305 K to 298 K in 7 cycles).
3.3.2. Microstructure of Ti-Ni-Hf-Cu-Nb Alloys

The backscatter images in Figure 8 show different microstructures of Ti-Ni-Hf-Cu-Nb alloys. Figure 8a shows the low $M_s$ sample has fine precipitates (5 to 10 μm) and smaller dendrite size (10 to 20 μm), while Figure 8b shows the high $M_s$ sample has coarse dendrites (20 to 40 μm) and several different types of precipitates of approximately 15 μm in size.

**Figure 7.** DSC results of (a) Ti$_{43}$Ni$_{33}$Hf$_{10.5}$Cu$_{11}$Nb$_{2.5}$ sample shows low $M_s$ (251 K), broad transformation peak (50 K) and good thermal cycle stability (three cycles overlapped); (b) Ti$_{44}$Ni$_{34}$Hf$_{10}$Cu$_{10}$Nb$_{2}$ sample shows high $M_s$ (305 K), narrow transformation peak (30 K) and weak thermal cycle stability.

The EDS area analysis results in Table 2 show the composition of these alloys were close to the target composition. EDS results of different phases are also included in Table 2. The Ti content in the matrix of the low $M_s$ sample is 2 at.% lower than that in the high $M_s$ sample, while the quantity of Hf in the matrix of low $M_s$ sample is 1 at.% higher. In addition, the Cu content in the low $M_s$ sample is 2 at.% greater than in the high $M_s$ sample. The total composition of Ti and Hf in the matrix of low $M_s$ sample is 50 at.%, 1 at.% lower than the high $M_s$.
sample. The Nb content in high $M_s$ sample is 0.3 at.\% higher. The lower Ti+Hf content and higher Cu+Nb content is one explanation as to why the low $M_s$ sample has a lower TT and higher thermal stability than the high $M_s$ sample. The results show that the distribution of elements in the matrix of these specified alloys has a strong influence on their $M_s$ and thermal stabilities.

Table 2. Chemical composition in Ti$_{43}$Ni$_{32}$Hf$_{10.5}$Cu$_{11.2}$Nb$_{2.5}$ and Ti$_{44}$Ni$_{34}$Hf$_{10}$Cu$_{10}$Nb$_2$ samples analysed by EDS.

| Sample                      | Low $M_s$ | High $M_s$ |
|-----------------------------|-----------|------------|
|                             | Ti$_{43}$Ni$_{32}$Hf$_{10.5}$Cu$_{11.2}$Nb$_{2.5}$ | Ti$_{44}$Ni$_{34}$Hf$_{10}$Cu$_{10}$Nb$_2$ |
| Matrix                      | Ti$_{38}$Ni$_{38}$Hf$_{11}$Cu$_{11}$Nb$_1$        | Ti$_{40.3}$Ni$_{38}$Hf$_{10.6}$Cu$_{10.4}$Nb$_{0.7}$ |
| Nb-rich (dark)              | Ti$_{60.2}$Ni$_{19}$Hf$_{5}$Cu$_{4.5}$Nb$_{21}$  | Ti$_{62.6}$Ni$_{19.5}$Hf$_{4.1}$Cu$_{2.7}$Nb$_{21.5}$ |
| Ti-rich (bright)            | Ti$_{57}$Ni$_{22}$Hf$_{5}$Cu$_{5}$Nb$_{2}$       | Ti$_{56.6}$Ni$_{17.2}$Hf$_{5.3}$Cu$_{5.8}$Nb$_{2.0}$ |
| (Ti+Cu)-rich (dark)         | -         | Ti$_{55.5}$Ni$_{17.0}$Hf$_{4.1}$Cu$_{16.2}$Nb$_{3.1}$ |
| Area average                | Ti$_{43.8}$Ni$_{34}$Hf$_{10}$Cu$_{5}$Nb$_{2.7}$  | Ti$_{43.1}$Ni$_{33}$Hf$_{10.5}$Cu$_{11}$Nb$_{2.5}$ |

The EDS mapping of the low $M_s$ sample in Figure 9 shows bright and dark contrasting precipitates within the sample, which are Ti-rich and Nb-rich, respectively. The matrix has the brightest contrast in BSE images and the EDS shows it had the highest Hf content, which suggests the most Hf was dissolved in the matrix. In the low $M_s$ sample, the brighter contrast Ti- and Hf-rich precipitate (indicated by red arrows in Figure 9) were more abundant, with 65 at.\% Ti+Hf content, which could be the (Ti, Hf)$_2$(Ni, Cu) precipitates with a small amount of Nb replacing both Ti/Hf and Ni/Cu. The Hf content in this precipitate is lower than that in the matrix but higher than the dark contrast precipitate. The darker contrast precipitates are low in Ni and Hf but rich in Nb, as indicated by the yellow arrows in Figure 9. The Nb-rich precipitates were found within and around the Ti-rich precipitates. The Nb content in this precipitate was greater than 20 at.\%, but the Ti+Hf content was still approximately 66 at.\%, which indicates this was (Ti, Hf)$_2$(Ni, Cu, Nb) with Nb replacing Ni and Cu. A similar microstructure in Ti$_{34.2}$Ni$_{17.8}$Hf$_{14}$Nb$_1$ SMA was reported by Lemke et al. [23], but the Ti was not homogeneous distributed in precipitates and the number of precipitates were less than the SDA Ti-rich SMAs. This is due to the higher Ti content, lower Nb content and higher cooling rate in Ti-rich specimen introduced more Ti-rich precipitates and enhance the homogenous precipitation.

In addition, the Cu elemental mapping in Figure 9 reveals segregation, with a moderately Cu-rich area within the matrix, near the precipitates, and a slightly Cu-lean area at the centre. As discussed in Section 3.1, Cu has a negative influence on $M_s$ but improves thermal cycle stability. Therefore, despite the $M_s$ in this sample being lower it was more stable than the high $M_s$ sample. The segregation of Cu also caused a broader transformation peak in Figure 7a. Yi et al. reported multi transformation peaks in the Ti-Ni-Hf SMAs with heterogeneous composition distribution [36].

Figure 10 shows the BSE image and EDS mapping of the high $M_s$ sample, which revealed three different precipitates. Those with the darkest contrast had two different compositions, (Ti+Cu)-rich (indicated by red arrows) and Nb-rich (indicated by white arrows). The average composition of the Ti+Cu-rich precipitates was Ti$_{55.3}$Ni$_{17.0}$Hf$_{8.1}$Cu$_{16.3}$Nb$_{1.1}$, which implied that Cu had replaced Ni and Nb replaced Ti/Hf in the (Ti, Hf)$_2$(Ni, Cu) precipitate. The average composition of the Nb-rich precipitate was Ti$_{62.6}$Ni$_{19.3}$Hf$_{1.3}$Cu$_{12.7}$Nb$_{21.3}$, close to that of the low $M_s$ sample. Some dendritic lacy $\beta$-Nb particles were found around the Ti-rich precipitates, which is similar to Lemke’s study with 8 at.\% Nb [23]. These precipitates also presented as small eutectic-like particles in the high $M_s$ sample.
Figure 9. EDS mapping around the precipitates in Ti$_{43}$Ni$_{33}$Hf$_{10.5}$Cu$_{11}$Nb$_{2.5}$ low Ms as-built sample shows high Nb content in the precipitates with the darkest contrast. The precipitates generally have higher Ti and Nb content.

The average composition of the Ti+Cu-rich precipitates was Ti$_{55.5}$Ni$_{17.0}$Hf$_{8.1}$Cu$_{16.3}$Nb$_{3.1}$, which implied that Cu had replaced Ni and Nb replaced Ti/Hf in the (Ti, Hf)$_2$(Ni, Cu) precipitate. The average composition of the Nb-rich precipitate was Ti$_{62.6}$Ni$_{9.3}$Hf$_{4.1}$Cu$_{2.7}$Nb$_{21.3}$, close to that of the low Ms sample. Some dendritic lacy β-Nb particles were found around the Ti-rich precipitates, which is similar to Lemke’s study with...
Figure 10. EDS mapping around the precipitates in Ti$_{44}$Ni$_{34}$Hf$_{10}$Cu$_{10}$Nb$_{2}$ high $M_s$ sample shows three different types of precipitates indicated by white, red and yellow arrows.

The brighter contrast precipitates (indicated by yellow arrows) were Ti-rich with the lowest Cu content. The average composition of this phase was Ti$_{56.6}$Ni$_{27.2}$Hf$_{8.3}$Cu$_{5.8}$Nb$_{2.0}$, similar to the Ti-rich precipitates in the low $M_s$ sample. The Cu-rich segregation area was smaller comparing to the low $M_s$ sample. There was a bright contrast area at the centre of the BSE image (indicated by a black arrow) with a composition of Ti$_{36.0}$Ni$_{33.9}$Hf$_{10.0}$Cu$_{15.5}$Nb$_{4.6}$, similar to the matrix composition (Ti$_{40.3}$Ni$_{38}$Hf$_{10.6}$Cu$_{10.4}$Nb$_{0.7}$) but with Cu replacing Ni and the Nb replacing Ti. This area would have had a much lower $M_s$ than the matrix but the volume was too small to be detected during the DSC test. Comparing
the results to Ti-Ni-Hf-Nb SMAs produced by arc melting [23,24], the higher Ti content in our study has a higher volume of precipitates and lower Nb content in the matrix, which indicates the Ti, Hf-rich precipitates absorbed Nb during solidification.

It could be that the Ti$_2$(Ni, Cu) particles reduced the stress field of the martensitic lattice and improved thermal-mechanical stability, which is consistent with the results reported by Chluba et al. [10]. However, the size of precipitates in bulk materials is much bigger than that in the thin film. Hence, the mechanism of improved stability remains further study.

3.3.3. Heat Treatment on Ti-Ni-Hf-Cu-Nb Samples

In order to homogenise the materials and improve shape memory properties, solution and homogenise treatments were applied to the Ti-Ni-Hf-Cu-Nb samples. The DSC results of the as-fabricated high $M_s$ sample seen in Figure 11 show that apart from the austenitic transformation peak at 300 K, four endothermic peaks occurred during heating. The first at 1174 K, slightly lower than the melting temperature of Ti$_2$Cu (1233 K) and Ti$_2$Ni (1239 K). This could have been the melting of the (Ti, Hf, Nb)$_2$(Ni, Cu) phase with a low Nb content and high Cu content, which was the bright contrast precipitates marked by red arrows in Figure 10. The second and third endothermic peaks occurred at 1448 K and 1563 K, respectively, possibly the melting of the matrix and the Cu segregation observed by EDS mapping. The final endothermic peak at 1755 K was perhaps the melting of the Nb-rich precipitates and β-Nb particles as Nb-Ni phase melting point exceed 1720 K when the Nb content is greater than 50 at.% [37].

![Graph showing DSC results](image)

**Figure 11.** High temperature DSC testing result of as-built Ti-Ni-Hf-Cu-Nb2.5 high $M_s$ sample indicates multi-phases and changes of local composition in some phases.

Three exothermic peaks emerged during cooling, with the first at 1719 K, 36 K lower than the last endothermic peak. They were the solidification and melting of the Nb-rich precipitates. The second and largest exothermic peak was at 1643 K. The magnified curve in Figure 12d reveals an additional emerged 1650 K peak. This might have been due to the diffusion and homogenisation of the chemical composition during the isothermal period at high temperature (1773 K). Since the peak width during cooling is much smaller than those of the heating peaks, the matrix was homogenised after the high temperature isothermal period. The third peak is much smaller than the other two, at 1324 K, which indicates
the volume of this phase was much smaller than the other. The third exothermic peak in Figure 12b is more than 100 K higher than the first endothermic peak during heating, which indicates the change of composition. According to the DSC result above, the low temperature precipitates melts below 1273 K while the matrix melt above this temperature. Therefore, a 1273 K three hours heat treatment was chosen for the homogenisation.

The backscatter image of the heat treated Ti-Ni-Hf-Cu-Nb low Ms sample shows the significantly different microstructure after heat treatment, as seen in Figure 13. The size of precipitates increased from 10 µm to more than 50 µm and a few different precipitates were formed after solution treatment. The EDS results in Figure 13 indicates that the precipitates at the edge with fine patterns (marked by yellow arrow) had a marginally higher Ti composition than the matrix, but lower than the other precipitate areas. The content of Ni, Hf, Cu elements in this phase was lower than the matrix area but the Nb content was higher. Considering the microstructure of these precipitates this could have been the globule β-Nb phase. The centre area of precipitates has higher Ti composition, lower Hf content but similar Nb content compared to the matrix. The areas with higher Ni content were marked by the white arrows and the areas with higher Cu content were marked by black arrows. Therefore, these phases may have been lower Nb content (Ti, Hf, Nb)2(Ni, Cu) phase with some Ni and Cu variation. The 2 to 3 µm precipitates with dark contrast (marked by red arrow) have high Ti and Nb composition but the lowest Ni, Cu, Hf content compared to the other phases, indicating this phase was (Ti, Hf, Nb)2(Ni, Cu) with high Nb content, which differs from the Ti2Ni type phases in Ti-Ni alloys. The phases marked by blue arrows have high Ti content and low Ni, Hf, Cu, Nb content, which indicates it is close to Ti2Ni precipitates.

In addition, there was no segregation found in the matrix, suggesting the homogenising treatment eliminated most of the segregation in Ti-Ni-Hf-Cu-Nb low Ms alloy. The DSC testing in Figure 14 shows that the transformation peak of heat treated sample was reduced to 20 K with the same thermal stability as the as-built sample, signifying that heat treatment can narrow the transformation range by eliminating segregation in the matrix. Although the high Ms sample was heat treated by the same process, the DSC result showed no significant difference to the as-built sample. With regard to different precipitates between as-built and heat treated samples, the better thermal cycle stability in the low Ms alloy might be due to the higher Cu content in the matrix as well as the interface between precipitates and matrix.

In order to investigate the segregation in the low Ms Ti-Ni-Hf-Cu-Nb alloy, a DSC sample of as-built condition and a heat treated DSC sample (HT3) were analysed by EDS. Three EDS mappings were applied on each sample at different positions, the top, middle and bottom areas of the samples. Each EDS mapping has 1024 × 800 index points and each point has 0.5 s index live time. The average composition of the matrix in each mapping area was calculated from manually selected data points where were at least 5 µm away from the precipitates, listed in Table 3. The results indicated that the compositional variation in the matrix of the as-built sample was larger than that of the heat treated sample. As discussed in previous sections, the reduction of Ti+Hf composition or increase of Nb content in the matrix leads to a lower transformation temperature in the DSC result. By using the equations in Section 2, and analysing the EDS result, the lowest Ms temperature of the as-built matrix was 25.3 K lower than the heat treated sample due to variation of Nb and Cu content, which was close to the experimental result (30 K). Therefore, the main reason for the broad transformation peak in the as-built low Ms sample was due to the segregation in the matrix, especially the heterogeneous distribution of Nb and Cu.
Figure 12. Magnified peak areas of high Ms Ti-Ni-Hf-Cu-Nb DSC curves (a) the endothermic/melting peak of the lowest melting phase during heating, (b) the exothermic/solidification peak of the lowest melting phase during cooling, (c) the endothermic/melting peak with the highest transformation energy phase during heating, (d) the exothermic/solidification peak with the highest transformation energy phase during cooling, (e) the endothermic/melting peak of the highest melting point phase during heating, (f) the exothermic/solidification peak with the highest melting point phase during cooling.
middle area of the as-fabricated sample. The backscatter image and EDS mapping of precipitates of heat treated Ti-Ni-Hf-Cu-Nb low Ms alloy showed no significant difference to the as-fabricated sample. Three EDS mappings were applied on each sample at different positions, the top, middle, and bottom areas of the samples. Each EDS mapping has 1024×1024 pixels. The average composition of the matrix in each mapping was calculated using the equations in Table 3. The results indicated that the main phases in the matrix are Nb and Cu with a composition close to Nb and Cu content, which was close to the Ti-Ni-Hf phase with some Ni and Cu variation. The 2 to 3 μm precipitates with dark contrast marked by the white arrows and the areas with higher Cu content were marked by black arrows. There were no significant differences in Nb and Cu content, which differs from the Ti-Ni-Hf-Cu-Nb alloy, a DSC testing result of heat treated Ti-Ni-Hf-Cu-Nb low Ms alloys show narrow transformation peaks and stable transformation behavior with similar transformation temperature to the as-fabricated sample.

Figure 13. Backscatter image and EDS mapping of precipitates of heat treated Ti-Ni-Hf-Cu-Nb low Ms sample shows the change of microstructure.

Figure 14. DSC testing result of heat treated Ti-Ni-Hf-Cu-Nb low Ms alloys show narrow transformation peaks and stable transformation behavior with similar transformation temperature to the as-fabricated sample.
Table 3. Table of average matrix composition calculated from EDS mapping of As-fabricated and heat treated low Ms Ti-Ni-Hf-Cu-Nb DSC samples (all composition in at.%).

| Sample  | Area  | Ti  | Ni  | Cu  | Nb  | Hf  | Ti+Hf | Ni+Cu |
|---------|-------|-----|-----|-----|-----|-----|-------|-------|
| As-built | Top   | 39.5| 36.9| 11.9| 1.4 | 10.2| 49.7  | 48.8  |
|         | Middle| 40.0| 36.3| 12.5| 1.0 | 10.3| 50.3  | 48.7  |
|         | Bottom| 40.1| 36.1| 12.6| 0.8 | 10.4| 50.5  | 48.7  |
|         | Min   | 39.3| 34.8| 11.4| 0.6 | 10.1| 49.5  | 47.9  |
|         | Max   | 40.4| 37.7| 13.7| 2.2 | 10.5| 50.8  | 49.3  |
| HT3     | Top   | 39.8| 36.5| 12.3| 1.1 | 10.4| 50.1  | 48.8  |
|         | Middle| 39.7| 36.6| 12.3| 1.0 | 10.3| 50.0  | 49.0  |
|         | Bottom| 40.0| 36.6| 12.3| 1.0 | 10.2| 50.2  | 48.8  |
|         | Min   | 39.5| 36.3| 11.6| 0.8 | 10.0| 49.7  | 48.0  |
|         | Max   | 40.2| 37.0| 12.8| 1.4 | 10.6| 50.3  | 49.2  |

3.3.4. Micromechanical Properties

In Figure 15, three pillars in the solution treated Ti-Ni-Hf-Cu-Nb low Ms samples were milled and compression tested at room temperature. The plateau phase occurred at approximately 700 MPa with the 2.5% strain, while the strain retrieved at 1500 MPa was 7.5%. The Young’s modulus before the plateau stage is 28 GPa and about 33 GPa after the plateau stage. The recoverable strain is approximately 3%. The Ti-Ni-Hf-Cu-Nb alloy has comparable plateau stress and irreversible deformations to the Ti-Ni alloys in similar testing conditions, but its elastic modulus was approximately 50% higher [38]. The Young’s modulus and recoverable strain of Ti-Ni-Hf-Cu-Nb alloys are similar to the Ti-Ni-Hf alloys (23 GPa before plateau stage and 3.1% recoverable strain) [8]. However, the thermal stability of Ti-Ni-Hf-Cu-Nb alloy is much better than the Ti-Ni-Hf alloys. This is due to the Cu improving the thermal stability via reducing the shear resistance, and the introduction of β-Nb particle increases the resistance of lattice deformation. The Nb particles and Ti/Hf-rich precipitates might also have pinning effect, which limits the accumulation of dislocation and increase the thermal stability.

![Figure 15. Micro-pillar compression test result of solution treated low Ms Ti-Ni-Hf-Cu-Nb sample.](image-url)
4. Conclusions

A novel Ti-Ni-Hf-Cu-Nb shape memory alloys with acceptable transformation temperatures and good thermal cycle stabilities have been developed. The influence of Cu, Nb and Hf on transformation temperature of Ti-Ni, Ti-Ni-Cu, Ti-Ni-Hf-Cu alloys have been investigated.

1. The addition of Cu in Ti-Ni SMAs slightly decreased the MT temperature by $-0.67 \, \text{K/at.\%}$.
2. Cu and Hf have a negative influence on each other on the MT temperature of Ti-Ni-Hf-Cu SMAs. The reduction of Cu on 10 at.% and 15 at.% Hf alloys were $-5.1 \, \text{K/at.\%}$ and $-5.7 \, \text{K/at.\%}$, respectively.
3. Nb has a stronger influence on the MT temperature of $\text{Ti}_{31-x}\text{Ni}_{39-x}\text{Cu}_{10}\text{Nb}_x$ shape memory alloys ($-19 \, \text{K/at.\%}$) than $\text{Ti}_{31-0.5x}\text{Ni}_{40-0.5x}\text{Nb}_x$ alloys ($-6.3 \, \text{K/at.\%}$). However, the Nb content did not affect the thermal cycle stability of $\text{Ti}_{31-x}\text{Ni}_{39-x}\text{Cu}_{10}\text{Nb}_x$ alloys when the Nb content is lower than 3 at.%. 
4. Nb content has less influence on MT temperature of $\text{Ti}_{41-0.5x}\text{Ni}_{46-0.5x}\text{Hf}_{10}\text{Cu}_{3x}\text{Nb}_x$ shape memory alloys ($-1.5 \, \text{K/at.\%}$) than $\text{Ti}_{41-0.5x}\text{Ni}_{39-0.5x}\text{Hf}_{10}\text{Cu}_{10}\text{Nb}_x$ alloys ($-12.5 \, \text{K/at.\%}$). However, the relationship between Nb content and the MT temperature of $\text{Ti}_{41-x}\text{Ni}_{39-x}\text{Hf}_{10}\text{Cu}_{10}\text{Nb}_{2x}$ alloys is not linear.
5. The $\text{Ti}_{41-0.5x}\text{Ni}_{39-0.5x}\text{Hf}_{10}\text{Cu}_{10}\text{Nb}_x$ alloys with 1 at.% Nb has the highest transformation temperature but the alloys with 2 at.% Nb has the best thermal cycle stability. The reduction of transformation heat and difference between austenitic transformation and MT increased with higher Nb content.
6. The $\text{Ti}_{31}\text{Ni}_{33}\text{Hf}_{10.5}\text{Cu}_{11}\text{Nb}_{2.5}$ and $\text{Ti}_{44}\text{Ni}_{33}\text{Hf}_{10}\text{Cu}_{10}\text{Nb}_2$ alloys showed completely different transformation behaviour due to different matrix composition. Three different precipitates were found in $\text{Ti}_{44}\text{Ni}_{33}\text{Hf}_{10}\text{Cu}_{10}\text{Nb}_2$ high Ms alloys which may be the reason for unstable transformation and the variation of transformation temperature.
7. The homogenised low Ms $\text{Ti}_{43}\text{Ni}_{33}\text{Hf}_{10.5}\text{Cu}_{11}\text{Nb}_{2.5}$ alloys has stable narrow transformation peak with no obvious thermal fatigue after a few thermal cycles. Three different types of precipitates were found in heat treated sample.
8. The micro-pillar compression test of low Ms showed two-stage deformation. Both the plateau stress and Young’s Modulus are higher than the Ti-Ni shape memory alloys.

Author Contributions: S.L.: Conceptualization, Writing—original draft; M.W.: Investigation, Data curation; K.E.: Methodology, Project administration; C.G.: Project administration; C.L.: Project administration; M.A.: Writing—review and editing, Supervision. All authors have read and agreed to the published version of the manuscript.

Funding: This work was financially supported by the [Guangzhou Science and Technology Plan Programme] (Project No. 201704030067) and was carried out in collaboration with Guangdong Academy of Sciences and the University of Birmingham. This project is also supported by the Accelerated Metallurgy Project, co-funded by the [European Commission in the seventh Framework Programme] (Contract NMP4-LA-2011-263206) by the [European Space Agency] by and the individual partner organisations.

Data Availability Statement: Not available.

Acknowledgments: The authors would like to thank Nicholas Adkins, who was a Senior Research Fellow in the University of Birmingham, providing helps on the experimental work, paper editing and project management of this work. The authors would also like to thank Yu-lung Chiu, Stephen McCain from University of Birmingham and Xiaohui Li from Guangdong Academy of Science providing helps on the experimental work and equipment.

Conflicts of Interest: The authors declare that they have no known competing financial interest or personal relationship that could have appeared to influence the work reported in this paper.

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