New Isocoumarin Analogues from the Marine-Derived Fungus Paraphoma sp. CUGBMF180003

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Abstract: Nine new secondary metabolites, including six isocoumarin analogues, 7-hydroxyoospolactone (1), 7-methoxyoospolactone (2), 7-methoxy-9-hydroxyoospolactone (3), 10-acetoxy-9-hydroxyoospolactone (4), 6-dehydroxysescandelin (5), parapholactone (6), and three compounds with a rare skeleton of isocoumarin coupled with phenylethylamine, namely paraphamide A (12), paraphamide B (13), and paraphamide C (14), together with five known compounds, oospolactone (7), 8-O-methyloospolactone (8), 10-hydroxyoospolactone (9), 9,10-dihydroxyoospolactone (10), and oospoglycol (11), were isolated and identified from the marine-derived fungus Paraphoma sp. CUGBMF180003. Their chemical structures were determined using spectroscopic data, including HRESIMS and 1D and 2D NMR techniques. Furthermore, the stereogenic carbons in 5 and 14 were determined by comparing the experimental and calculated electronic circular dichroism (ECD) spectra. The carbon skeleton of 12–14 was identified as the first example of isocoumarin coupled with phenylethylamine derivatives. All of these compounds were examined for antimicrobial activities against Candida albicans and Staphylococcus aureus. Both 1 and 6 showed antibacterial activity against S. aureus with MIC values of 12.5 µg/mL.

Keywords: marine-derived fungus; Paraphoma sp.; natural products; isocoumarin derivatives; antibacterial activity

1. Introduction

The Paraphoma fungi are commonly isolated from soil samples or plants and recognized as soil-borne pathogens [1,2]. Strains belonging to this genus have been proven to degrade plastic films such as poly(butylene succinate-co-butylene adipate) and poly(butylene succinate) [3,4]. Some secondary metabolites, such as polyketides [5], phenol, latam [6], and isochromenones [7] have been identified in fungi of this genus. Our present chemical investigation on the marine-derived fungus Paraphoma sp. strain CUGBMF180003, isolated from a mud sample collected from Shenzhen, China, led to the identification of nine new secondary metabolites, including six isocoumarin analogues, namely, 7-hydroxyoospolactone (1), 7-methoxyoospolactone (2), 7-methoxy-9-hydroxyoospolactone (3), 10-acetoxy-9-hydroxyoospolactone (4), 6-dehydroxysescandelin (5), parapholactone (6), three compounds with a rare skeleton of isocoumarin coupled with phenylethylamine, namely, paraphamide A (12), paraphamide B (13), and paraphamide C (14), as well
as five known compounds, oospolactone (7) [8], 8-O-methoxyospolactone (8) [9], 10-hydroxyospolactone (9) [10], 9,10-dihydroxyospolactone (10) [10], and oospoglycol (11) [11]. Herein, we report the isolation, structure elucidation, and biological activities of these compounds.

2. Results
2.1. Structure Elucidation

Compound 1 was isolated as a light yellow powder. The molecular formula of 1 was determined to be C_{11}H_{10}O_{4} based on the HRESIMS spectrum (m/z [M + H]^+ 207.0652, calcd. for C_{11}H_{11}O_{4} 207.0652), accounting for seven degrees of unsaturation (Figure S1). The \(^1\)H NMR data of 1 (Table 1, Figure S2) demonstrated two aromatic proton signals ([\(\delta_H\) 6.93, d, \(J = 8.5\) Hz, H-5), (\(\delta_H\) 7.31, 1H, d, \(J = 8.5\) Hz, H-6)], two methyl singlets at \(\delta_H\) 2.24 (3H, s, H-9) and 2.06 (3H, s, H-10), and two phenolic hydroxyl groups at \(\delta_H\) 9.69 (1H, brs, 7-OH) and 11.12 (1H, brs, 8-OH). The \(^13\)C and HSQC spectra of 1 (Figures S3 and S4) indicated 11 carbon signals (Table 1) which were categorized as one carboxyl at \(\delta_C\) 166.2 (C-1), three oxygenated sp\(^2\) carbons at \(\delta_C\) 146.7 (C-3), 143.8 (C-7), and 148.6 (C-8), two nonprotonated sp\(^2\) carbons at \(\delta_C\) 129.6 (C-4a) and 105.9 (C-8a), and two protonated sp\(^2\) carbons at \(\delta_C\) 113.3 (C-5) and 124.2 (C-6). Detailed analysis of the 2D NMR data (Figures S4–S6) revealed the existence of an isocoumarin skeleton for 1. The HMBC correlations from H-5 to C-7 and C-8a, from H-6 to C-4a, C-7 and C-8 revealed the o-dihydroxy substituted benzene moiety. The long-range HMBC correlation from H-5 to C-1 indicated the presence of a carboxyl substituent on C-8a. Furthermore, the HMBC correlations from H-9 and H-10 to C-3 and C-4 indicated that the methyl groups were on C-3 and C-4. The connection from C-4 to C-4a was evidenced by the HMBC correlation from H-10 to C-4a. With the downfield chemical shifts of C-1 and C-3, it was revealed that C-1 and C-3 were connected by an oxygen atom. Therefore, the structure of 1 was assigned as shown in Figure 1 and named 7-hydroxyospolactone.

Figure 1. Chemical structures of 1–14.

Compound 2 was isolated as a light yellow powder. The molecular formula of 2 was determined to be C_{12}H_{13}O_{4} based on the HRESIMS spectrum (m/z [M + H]^+ 221.0811, calcd. for C_{12}H_{13}O_{4} 221.0808), accounting for seven degrees of unsaturation (Figure S7). The \(^1\)H NMR data of 2 (Table 1, Figure S8) displayed similar signals to those of 1. Comparison of the NMR data of the two compounds revealed the presence of one methoxyl signal at \(\delta_H\) 3.85, s (\(\delta_C\) 56.1) in 2. The HMBC spectrum (Figure S12) exhibited correlations from H-5 and -OMe to C-7, indicating that the methoxyl group was on C-7. Therefore, the structure of 2 was elucidated as shown in Figure 1 and named 7-methoxyospolactone.
Table 1. $^1$H (500 MHz) and $^{13}$C NMR (125 MHz) NMR data of 1–3.

| Position | 1 (DMSO-d$_6$) | δ$^1$H (J in Hz) | δ$^{13}$C, Type | 2 (DMSO-d$_6$) | δ$^1$H (J in Hz) | δ$^{13}$C, Type | 3 (DMSO-d$_6$) | δ$^1$H (J in Hz) |
|----------|----------------|-----------------|----------------|----------------|-----------------|----------------|----------------|-----------------|
| 1        | 166.2, CO      | 166.2           | 166.2          | 166.2          | 166.2           | 166.2          | 166.2          | 166.2           |
| 2        | 148.6, C       | 150.2           | 149.5          | 150.2          | 150.2           | 150.2          | 150.2          | 150.2           |
| 3        | 145.7, C       | 146.4           | 110.5          | 150.2          | 150.2           | 150.2          | 150.2          | 150.2           |
| 4        | 143.8, C       | 130.4           | 130.0          | 150.5          | 150.5           | 150.5          | 150.5          | 150.5           |
| 4a       | 129.6, C       | 113.0           | 113.9          | 120.4          | 121.4           | 121.4          | 121.4          | 121.4           |
| 5        | 124.2, CH      | 120.6           | 120.4          | 120.4          | 120.4           | 120.4          | 120.4          | 120.4           |
| 6        | 143.8, C       | 113.0           | 113.9          | 113.9          | 113.9           | 113.9          | 113.9          | 113.9           |
| 7        | 145.7, C       | 120.6           | 120.4          | 120.4          | 120.4           | 120.4          | 120.4          | 120.4           |
| 8        | 146.6, C       | 113.0           | 113.9          | 113.9          | 113.9           | 113.9          | 113.9          | 113.9           |
| 8a       | 105.9, C       | 105.6           | 105.6          | 105.6          | 105.6           | 105.6          | 105.6          | 105.6           |
| 9        | 16.5, CH$_3$   | 16.5            | 57.6           | 16.5           | 16.5            | 57.6           | 16.5           | 16.5            |
| 10       | 12.0, CH$_3$   | 11.9            | 11.5           | 11.9           | 11.9            | 11.5           | 11.9           | 11.9            |
| 7-OH/-  | OMe            | 9.69, br s       | 5.16           | 9.69           | 9.69            | 5.16           | 9.69           | 9.69            |
| 8-OH    | 11.12, s       | 11.24, s        | 11.24, s       | 11.24          | 11.24           | 11.24          | 11.24          | 11.24           |

Compound 3 was isolated as a light yellow powder. The molecular formula of 3 was determined to be C$_{12}$H$_{12}$O$_3$ based on the HRESIMS spectrum ($m/z$ [M + H]$^+$ 237.0758, calcd. for C$_{12}$H$_{12}$O$_3$: 237.0757), accounting for seven degrees of unsaturation (Figure S13). The $^1$H NMR data of 3 (Table 1, Figure S14) displayed similar signals to those of 2, except for the presence of the signal of hydroxymethyl protons δ$_H$ 4.36 (2H, s, H-9), δ$_C$ 57.6 C-9). Detailed analysis of $^{13}$C and 2D NMR data (Figures S15–S17) revealed the structure of 3. The HMBC spectrum (Figure 2, Figure S17) exhibited correlations from H-9 to C-3 and C-4, and from H-10 to C-3, C-4, and C-4a indicating that the hydroxymethyl group was on C-3. Therefore, the structure of 3 was elucidated as shown in Figure 1 and named 7-methoxy-9-hydroxyoospolactone.

Figure 2. Key COSY (bold lines) and HMBC (arrows) correlations in 1–6 and 12–14.

Compound 4 was isolated as a light yellow powder. The molecular formula of 4 was determined to be C$_{13}$H$_{12}$O$_6$ based on the HRESIMS spectrum ($m/z$ [M + H]$^+$ 265.0708, calcd. for C$_{13}$H$_{12}$O$_6$: 265.0707), accounting for eight degrees of unsaturation (Figure S18). The $^1$H NMR data of 4 (Table 2, Figure S19) displayed signals of three aromatic protons at δ$_H$ 7.33 (1H, d, $J = 8.0$ Hz, H-5), 7.70 (1H, dd, $J = 8.0$, 8.0 Hz, H-6), and 7.06 (1H, d, $J = 8.0$ Hz, H-7), two hydroxymethyl protons at δ$_H$ 4.82 (2H, s, H-9), and two oxymethylene protons at δ$_H$ 5.09 (2H, s, H-10), as well as one methyl singlet at δ$_H$ 2.12 (3H, s, H-12). The $^{13}$C and HSQC spectra of 4 (Figures S20 and S21) showed 13 carbon signals, including...
those of the isocoumarin skeleton similar to those of oospolactone, and the two methyls in oospolactone were replaced by two hydroxymethyls at δ C 57.1 (C-9) and 60.1 (C-10), as well as two signals for acetyl groups at δ C 171.8 (C-11) and 21.0 (C-12). The presence of the hydroxymethyl group on C-3 was confirmed by the HMBC correlations (Figure S23) from H-9 to C-3 and C-4. Additionally, the HMBC correlations from H-10 to C-3, C-4, C-4a, and C-11 and from H-12 to C-11 revealed an acetoxy group on C-10. Thus, the structure of 4 was elucidated as shown in Figure 1 and named 10-acetoxy-9-hydroxyoospolactone.

Table 2. $^1$H (500 MHz) and $^{13}$C NMR (125 MHz) NMR data of 4–6.

| Position | 4 (CDCl$_3$) | 5 (CDCl$_3$) | 6 (Acetone-$d_6$) |
|----------|--------------|--------------|-----------------|
|          | $\delta$C, Type | $\delta$H ($f$ in Hz) | $\delta$C, Type | $\delta$H ($f$ in Hz) | $\delta$C, Type | $\delta$H ($f$ in Hz) |
| 1        | 165.6, CO     | 166.4, CO    | 163.2, CO       |
| 3        | 148.1, C      | 141.2, CH    | 7.37, s         | 151.1, C      |
| 4        | 117.7, C      | 122.1, C     | 127.3, C        |
| 4a       | 136.5, C      | 135.7, C     | 133.9, C        |
| 5        | 114.9, CH     | 7.33, d (8.0) | 113.4, CH      | 7.13, d (8.0) | 118.9, CH      | 7.50, dd (7.5, 1.0) |
| 6        | 137.8, CH     | 7.70, dd (8.0, 8.0) | 137.4, CH   | 7.66, dd (8.0, 8.0) | 137.5, CH   | 7.69, dd (8.5, 7.5) |
| 7        | 116.7, CH     | 7.06, d (8.0) | 116.0, CH     | 7.02, d (8.0) | 119.0, CH     | 7.94, dd (8.5, 1.0) |
| 8        | 162.1, C      | 106.8, C     | 136.2, C       |
| 8a       | 106.9, C      | 106.8, C     | 107.8, C       |
| 9        | 57.1, CH$_2$  | 4.82, s      | 65.0, CH      | 5.06, q (6.5) | 75.3, CH      | 5.83, dd (6.0, 3.0) |
| 10       | 60.1, CH$_2$  | 5.09, s      | 23.3, CH$_3$  | 1.60, d (6.5) | 39.9, CH$_2$  | 2.76, dd (18.0, 3.0) |
| 11       | 171.8, C      | 11.06, s     | 11.28, s       |
| 12       | 21.0, CH$_3$  | 2.12, s      |                  |
| 8-OH     | 11.06, s     | 11.28, s     |                  |

Compound 5 was isolated as a light yellow powder. The molecular formula of 5 was determined to be C$_{11}$H$_{10}$O$_4$ based on the HRESIMS spectrum ($m/z$ [M $+$ H]$^+$ 207.0648, calcd. for C$_{11}$H$_{11}$O$_4$ 207.0652), accounting for seven degrees of unsaturation (Figure S24). The $^1$H data of 5 (Table 2, Figure S25) displayed three aromatic proton signals at $\delta$H 7.13 (1H, d, $J$ = 8.0 Hz, H-5), 7.66 (1H, dd, $J$ = 8.0, 8.0 Hz, H-6), and 7.02 (1H, d, $J$ = 8.0 Hz, H-7), one olefinic proton signal at 7.37 (1H, s, H-3), one methyl doublet at $\delta$H 1.60 (3H, d, $J$ = 6.5 Hz, H-10), and one oxygenated sp$^3$ methine at $\delta$H 5.06 (1H, q, $J$ = 6.5 Hz, H-9). The $^{13}$C and HSQC spectra (Figures S26 and S27) presented 11 carbon resonances, including one carboxyl at $\delta$C 166.4 (C-1), four protonated sp$^2$ at $\delta$C 141.2 (C-3), 113.4 (C-5), 137.4 (C-6), 116.0 (C-7), four nonprotonated sp$^2$ carbon signals at $\delta$C 122.1 (C-4), 135.7 (C-4a), 162.4 (oxygenated, C-8), and 106.8 (C-8a), as well as one signal for oxygenated methine sp$^3$ at $\delta$C 65.0 (C-9) and one methyl carbon at $\delta$C 23.3 (C-10). Detailed analysis of the 2D NMR data (Figures S27–S29) revealed that 5 is an analogue of the isocoumarin oospolactone. However, the methyl group at C-3 was replaced by a proton, and another methyl group at C-4 was replaced by a 1-hydroxyethyl group. The proposed structure was confirmed by the HMBC correlations from H-3 to C-1, C-4a, C-9, and C-10, and from H-10 to C-4 and C-9. Thus, the planar structure of 5 was elucidated. The absolute configuration of C-9 was assessed by comparison of experimental and calculated ECD spectra (Figure 3A), confirming the S configuration of C-9, and the compound was named 6-dehydroxysescandelin.
Compound 6 was isolated as a light yellow powder. The molecular formula of 6 was determined to be C_{12}H_{10}O_{5} based on the HRESIMS spectrum (m/z [M + H]^+ 233.0443, calcd. for C_{12}H_{10}O_{5}, 233.0444), accounting for nine degrees of unsaturation (Figure S30). The \(^1\)H NMR data of 6 (Table 2, Figure S31) displayed signals of three aromatic protons at δ\(_{\text{H}}\) 7.50 (1H, dd, J = 7.5, 1.0 Hz, H-5), 7.69 (1H, dd, J = 8.5, 7.5 Hz, H-6), and 7.04 (1H, dd, J = 8.5, 1.0 Hz, H-7), two methylene sp\(^3\) protons at δ\(_{\text{H}}\) 3.06 (1H, dd, J = 18.0, 6.0 Hz, H-10a) and 2.76 (1H, dd, J = 18.0, 3.0 Hz, H-10b), and one oxygenated methine sp\(^3\) proton at δ\(_{\text{H}}\) 5.83 (1H, dd, J = 6.0, 3.0 Hz, H-9). The \(^{13}\)C NMR and HSQC spectra (Figures S32 and S33) revealed twelve carbon resonances including one ketone carbonyl at δ\(_{\text{C}}\) 196.4 (C-11), one carboxyl at δ\(_{\text{C}}\) 163.2 (C-1), three protonated sp\(^2\) at δ\(_{\text{C}}\) 188.9 (C-5), 137.5 (C-6), 119.0 (C-7), three nonprotonated sp\(^2\) at δ\(_{\text{C}}\) 127.3 (C-4), 133.9 (C-4a), and 107.8 (C-8a), two oxygenated sp\(^2\) at δ\(_{\text{C}}\) 151.1 (C-3) and 163.2 (C-8), and one methylene sp\(^3\) at δ\(_{\text{C}}\) 39.9 (C-10) and one oxymethine sp\(^3\) at δ\(_{\text{C}}\) 75.3 (C-9). Detailed analysis of the 2D NMR data of 6 (Figures S33–S35) revealed an isocoumarin analogue. HMBC correlations from H-10 to C-3, C-4, and C-11, and from H-9 to C-3 and C-4 suggested that the 4-hydroxycyclopentenone ring was fused with the chromone ring through C-3 and C-4. Compound 6 showed weak optical rotation value of +0.91 (c 0.11, MeOH) and did not display ECD absorptions; therefore, 6 was elucidated as a racemic mixture. Therefore, the structure of 6 was defined as shown in Figure 1 and named paraphamide A.

Compound 12 was isolated as a light yellow powder. The molecular formula of 12 was determined to be C_{19}H_{15}NO_{4} based on the HRESIMS spectrum (m/z [M + H]^+ 322.1069, calcd. for C_{19}H_{15}NO_{4}, 322.1074), accounting for thirteen degrees of unsaturation (Figure S36). The \(^1\)H and \(^{13}\)C NMR data of 12 (Table 3, Figures S37 and S38) displayed proton and carbon signals of isocoumarin moiety similar to those of oospolactone, as well as resonances for one phenethyl group [δ\(_{\text{H}}\) 3.73, t, J = 7.5 Hz, H-1', δ\(_{\text{C}}\) 43.7 C-1'; δ\(_{\text{H}}\) 2.93, t, J = 7.5 Hz, H-2', δ\(_{\text{C}}\) 33.8 C-2'; δ\(_{\text{H}}\) 7.27, d, J = 7.0 Hz, H-4'/8'; δ\(_{\text{C}}\) 128.6 C-4'/8'; δ\(_{\text{H}}\) 7.30, dd, J = 7.0, 7.0 Hz, H-5'/7', δ\(_{\text{C}}\) 128.4 C-5'/7'; δ\(_{\text{H}}\) 7.21, t, J = 7.0 Hz, H-6', δ\(_{\text{C}}\) 126.3 C-6'], one methylene at δ\(_{\text{H}}\) 4.48 (s, H-9), δ\(_{\text{C}}\) 45.6 (C-9) and one conjugated amide carbonyl at δ\(_{\text{C}}\) 160.9 (C-10). Detailed analysis of 2D NMR data (Figures S39–S41) confirmed the connections of the phenethyl moiety to the isocoumarin core. The molecular formula (C_{19}H_{15}NO_{4}) of 12 and downfield shift of C-9 (δ\(_{\text{C}}\) 45.6), combined with the HMBC correlations from H-9 to C-3, C-4, C-10, and C-1', and from H-1' to C-9 and C-10 confirmed the connection of C-1', C-9, and C-10 to N, forming a cyclopentenamide ring. Therefore, the structure of 12 was elucidated as shown in Figure 1 and named parapholactone A.
| Position | δ_C (DMSO-d6) | δ_H (J in Hz) | δ_C (DMSO-d6) | δ_H (J in Hz) | δ_C (DMSO-d6) | δ_H (J in Hz) |
|----------|--------------|----------------|--------------|----------------|--------------|----------------|
| 1        | 164.4, CO    |                | 164.4, CO    |                | 165.4, CO    |                |
| 3        | 143.5, C     |                | 143.6, C     |                | 143.7, C     |                |
| 4        | 122.9, C     |                | 122.9, C     |                | 123.7, C     |                |
| 4a       | 132.4, C     |                | 132.4, C     |                | 132.2, C     |                |
| 5        | 113.7, CH    | 7.11, d (8.0)  | 113.8, CH    | 7.11, d, 8.0   | 113.0, CH    | 6.88, d (7.5)  |
| 6        | 138.1, CH    | 7.82, dd (8.0, 8.0) | 138.1, CH | 7.81, dd, 8.0, 8.0 | 138.0, CH | 7.68, dd (8.0, 7.5) |
| 7        | 117.5, CH    | 7.14, d (8.0)  | 117.5, CH    | 7.14, d, 8.0   | 118.6, CH    | 7.12, d (8.0)  |
| 8        | 161.4, C     |                | 161.4, C     |                | 163.0, C     |                |
| 8a       | 106.4, C     |                | 106.5, C     |                | 106.8, C     |                |
| 9        | 45.6, CH₂    | 4.48, s        | 45.6, CH₂    | 4.45, s        | 43.7, CH₂    | 4.57, d (17.5) |
| 10       | 160.9, CO    |                | 160.8, CO    |                | 162.1, CO    |                |
| 1'       | 43.7, CH₂    | 3.73, t (7.5)  | 44.1, CH₂    | 3.66, t, 7.5   | 55.0, CH     | 5.34, dd (10.5, 5.5) |
| 2'       | 33.8, CH₂    | 2.93, t (7.5)  | 33.0, CH₂    | 2.80, t (7.5)  | 35.9, CH₂    | 3.51, dd (14.5, 5.5) |
| 3'       | 138.6, CH    |                | 128.6, C     |                | 135.8, C     |                |
| 4'       | 128.6, CH    | 7.27, d (7.0)  | 129.5, CH    | 7.04, d (8.5)  | 128.5, CH    | 7.21, d (8.0)  |
| 5'       | 128.4, CH    | 7.30, dd (7.0, 7.0) | 115.3, CH | 6.67, dd (8.5) | 129.1, CH | 7.28, dd (8.0, 8.0) |
| 6'       | 126.3, CH    | 7.21, t (7.0)  | 155.8, C     |                | 127.4, CH    | 7.22, t (8.0)  |
| 7'       | 128.4, CH    | 7.30, dd (7.0, 7.0) | 115.3, CH | 6.67, d (8.5)  | 129.1, CH | 7.28, dd (8.0, 8.0) |
| 8'       | 128.6, CH    | 7.27, d (7.0)  | 129.5, CH    | 7.04, d (8.5)  | 128.5, CH    | 7.21, d (8.0)  |
| 1"       |                |                | 171.1, CO    |                |              |                |
| 2"       |                |                | 52.9, CH₃    | 3.76, s        |              |                |

Compound 13 was isolated as a light yellow powder. The molecular formula of 13 was determined to be C₁₉H₁₅NO₅ based on the HRESIMS spectrum (m/z [M + H]^+ 338.1022, calcd. for C₁₉H₁₆NO₅, 338.1023), accounting for thirteen degrees of unsaturation (Figure S42). The $^1$H and $^{13}$C NMR spectra of 13 (Table 3, Figures S43 and S44) resembled those of 12. Detailed analysis of the $^1$H and $^{13}$C NMR data (Figures S43 and S44) revealed that H-6’ of 12 was replaced by a hydroxyl group in 13. The structure of 13 was confirmed by the downfield shift of C-6’ (δ_C 155.8) and molecular formula. Thus, the structure of 13 was elucidated as shown in Figure 1 and named paraphamide B.

Compound 14 was isolated as a light yellow powder. The molecular formula of 14 was determined to be C₂₁H₁₇NO₆ based on the HRESIMS spectrum (m/z [M + H]^+ 380.1130, calcd. for C₂₁H₁₈NO₆, 380.1129), accounting for fourteen degrees of unsaturation (Figure S48). The $^1$H and $^{13}$C NMR spectra of 14 (Table 3, Figures S49 and S50) resembled those of 12. Detailed analysis of 2D NMR spectra (Figures S51–S53) revealed that one of the protons attached to C-1’ of 12 was replaced by a methyl formate group with resonances at δ_C 171.1 (C-1”), δ_H 3.76 (3H, s, H-2”), and δ_C 52.9 (C-2”). This moiety was confirmed by HMBC correlations from H-1’’, H-2’’, and H-2” to C-1”. The absolute configuration of C-1’’ was also determined by comparison of experimental and calculated ECD spectra (Figure 3B), confirming the $S$ configuration of C-1’’. Thus, the structure of 14 was elucidated as shown in Figure 1 and named as paraphamide B.

Five known oospolactone analogues were isolated from Paraphoma sp. CUGBMF180003 and identified as oospolactone (7) [8], 8-O-methylloospolactone (8) [9], 10-hydroxyoospolactone (9) [10], 9,10-dihydroxyoospolactone (10) [10], and oospoglycol (11) [11], by comparing their spectroscopic data with the respective previously reported data.

### 2.2. Biological Activity

All of the isolated compounds were subjected to tests of antibacterial activities against Candida albicans ATCC 10231 and Staphylococcus aureus ATCC 25923. Both 1 and 6 showed...
an inhibitory effect against *S. aureus* with minimum inhibitory concentration (MIC) values of 12.5 µg/mL; however, none of the isolates inhibited the growth of *C. albicans*.

3. Materials and Methods

3.1. General Experimental Procedures

Optical rotations ([α]$_{25}^{D}$) were measured on an Anton Paar MCP 200 Modular Circular Polarimeter (Anton Paar GmbH, Graz, Austria) in a 100 × 2 mm cell. CD spectra were recorded on an Applied Photophysics Chirascan spectropolarimeter (Applied Photophysics Ltd., Leatherhead, UK). NMR spectra were obtained on a Bruker Avance 500 spectrometer (Bruker BioSpin Corp., Billerica, MA, USA) with residual solvent peaks as references (DMSO-$d_6$: δ$_H$ 2.50, δ$_C$ 39.52; acetone-$d_6$: δ$_H$ 2.05, δ$_C$ 29.84; CDCl$_3$: δ$_H$ 7.26, δ$_C$ 77.16). High-resolution ESIMS measurements were obtained on an Accurate-Mass-Q-TOF LC/MS 6520 instrument (Agilent Technologies, Santa Clara, CA, USA) in positive ion mode. HPLC was performed using an Agilent 1200 Series separation module equipped with an Agilent 1200 Series diode array and Agilent 1260 Series fraction collector (Agilent Technologies, Santa Clara, CA, USA), and an Agilent ZORBAX SB-C18 column (250 × 9.4 mm, 5 µm).

3.2. Microbial Material, Fermentation, Extraction, and Purification

Strain CUGBMF180003 was isolated from a mud sample collected from the intertidal zones of Shenzhen, China, and grown on a potato dextrose agar plate at 28 °C. The genomic DNA of CUGBMF180003 was extracted using the GO-GPLS-100 kit (GeneOn BioTech, Changchun, China). The ITS region was amplified using a conventional primer pair of ITS4 (5′-TCTTCCGCTTATTGATATGC-3′) and ITS5 (5′-GGAAGTAAAAGTCGTAACAAGG-3′). PCR products were sent to Beijing Qingke Biotechnology Co., Ltd. (Beijing, China) for DNA sequencing and deposited in GenBank (accession number, MZ268156). CUGBMF180003 was identified as *Paraphoma* sp. by sequence analysis of its internal transcribed spacer (ITS) region and comparison with sequences from the GenBank database, using the BLAST program to determine an approximate phylogenetic affiliation. Alignments and calculations of sequence similarity were carried out using CLUSTAL W [12]. A neighbor-joining (NJ) tree (Figure S54) was constructed using the software package Mega version 5 [13]. Bootstrap resampling method with 1000 replicates was used in evaluating the topology of the phylogenetic trees [14]. The fungus was assigned the accession number CUGBMF180003 in the culture collection at the China University of Geosciences, Beijing. The strain CUGBMF180003 was inoculated on a potato dextrose agar plate and cultured for 5 days. Subsequently, a slit of agar with fungus was cut from the plate and inoculated into 10 1 L conical flasks, each containing solid medium consisting of rice (200 g) and artificial seawater (3.5%; 200 mL), and the flasks were incubated under static conditions at 28 °C for 30 days. The cultures were extracted three times with a mixture of EtOAc:MeOH (80:20), and the combined extracts were evaporated to dryness in vacuo. The residue was suspended in distilled water and partitioned with EtOAc. The EtOAc layer was then dried in vacuo to yield a dark residue (17.25 g). The EtOAc fraction was fractionated via vacuum liquid silica gel chromatography (80 × 80 mm column, silica gel 60 H for thin-layer chromatography) using a stepwise gradient of 80–100% hexane/CH$_2$Cl$_2$ and then 0–90% MeOH/CH$_2$Cl$_2$ to afford 12 fractions. Fraction C was fractionated on a Sephadex LH-20 column using an isocratic elution of CH$_2$Cl$_2$:MeOH (2:1) to yield six subfractions (C1–C6), and subfraction C6 was further fractionated by HPLC (Agilent ZORBAX SB-C18, 250 × 9.4 mm, 5 µm column, 3.0 mL/min, elution with 40% to 100% acetonitrile/H$_2$O) to yield 7 (46.9 mg). Fraction D was fractionated on a Sephadex LH-20 column using an isocratic elution of CH$_2$Cl$_2$:MeOH (2:1) to give four subfractions (D1–D4). Subfraction D3 was further fractionated by HPLC (Agilent ZORBAX SB-C18, 250 × 9.4 mm, 5 µm column, 3.0 mL/min, elution with 40% to 100% acetonitrile/H$_2$O) to yield 1 (9.5 mg) and 2 (1.3 mg). Fraction F was fractionated on a Sephadex LH-20 column using an isocratic elution of CH$_2$Cl$_2$:MeOH (2:1) to give six subfractions (F1–F6). Subfraction F3 was further fractionated by HPLC (Agilent ZORBAX SB-C18, 250 × 9.4 mm, 5 µm column, 3.0 mL/min, with 30% to 70% acetonitrile/H$_2$O) to
yield 8 (3.1 mg), 12 (7.5 mg), and 14 (1.2 mg). Subfraction F5 was further fractionated by HPLC (Agilent ZORBAX SB-C18, 250 × 9.4 mm, 5 µm column, 3.0 mL/min, elution with 30% to 60% acetonitrile/H₂O) to yield 4 (2.7 mg), 5 (2.3 mg), 6 (8.7 mg), and 9 (10.8 mg). Fraction M was fractionated on a Sephadex LH-20 column using an isocratic elution of CH₂Cl₂:MeOH (2:1) to give nine subfractions (M1–M9). Subfraction M7 was further fractionated by HPLC (Agilent ZORBAX SB-C18, 250 × 9.4 mm, 5 µm column, 3.0 mL/min, elution with 30% to 72% acetonitrile/H₂O) to yield 3 (6.3 mg) and 13 (5.8 mg). Fraction N was fractionated on a Sephadex LH-20 column using an isocratic elution of CH₂Cl₂:MeOH (2:1) to give seven subfractions (N1–N7). Subfraction N7 was further fractionated by HPLC (Agilent ZORBAX SB-C18, 250 × 9.4 mm, 5 µm column, 3.0 mL/min, elution with 30% to 72% acetonitrile/H₂O) to yield 10 (1.7 mg) and 11 (6.3 mg).

7-Hydroxyoospolactone (1): Light yellow powder; ¹H and ¹³C NMR data, Table 1; HRESIMS m/z 207.0652 [M + H]⁺ (calcd. C₁₁H₁₁O₄, 207.0652).

7-Methoxyoospolactone (2): Light yellow powder; ¹H and ¹³C NMR data, Table 1; HRESIMS m/z 221.0811 [M + H]⁺ (calcd. for C₁₂H₁₃O₄, 221.0808).

7-Methoxy-9-hydroxyoospolactone (3): Light yellow powder; ¹H and ¹³C NMR data, Table 1; HRESIMS m/z 241.0705 [M + H]⁺ (calcd. for 237.0758, C₁₂H₁₃O₃, 237.0757).

10-Acetoxy-9-hydroxyoospolactone (4): Light yellow powder; ¹H and ¹³C NMR data, Table 2; HRESIMS m/z 265.0708 [M + H]⁺ (calcd. for C₁₃H₁₃O₆, 265.0707).

6-Dehydroxyssescandelin (5): Light yellow powder; [α]D²⁰ −16.5 (c 0.20, MeOH); ¹H and ¹³C NMR data, Table 2; HRESIMS m/z 207.0648 [M + H]⁺ (calcd. for C₁₁H₁₁O₄, 207.0652).

Parapholactone (6): Light yellow powder; [α]D²⁰ +0.91 (c 0.11, MeOH); ¹H and ¹³C NMR data, Table 2; HRESIMS m/z 233.0443 [M + H]⁺ (calcd. for C₁₂H₉O₅, 233.0444).

Paraphamide A (12): Light yellow powder; ¹H and ¹³C NMR data, Table 3; HRESIMS m/z 322.1069 [M + H]⁺ (calcd. for C₁₉H₁₆NO₄, 322.1074).

Paraphamide B (13): Light yellow powder; ¹H and ¹³C NMR data, Table 3; HRESIMS m/z 338.1022 [M + H]⁺ (calcd. for C₁₉H₁₈NO₅, 338.1023).

Paraphamide C (14): Light yellow powder; [α]D²⁰ +35.0 (c 0.06, MeOH); ¹H and ¹³C NMR data, Table 3; HRESIMS m/z 380.1130 [M + H]⁺ (calcd. for C₂₁H₁₈NO₆, 380.1129).

### 3.3. Biological Activity

Compounds 1–14 were evaluated for their antimicrobial activities in 96-well plates according to the antimicrobial susceptibility testing standards outlined by the Clinical and Laboratory Standards Institute document M07-A7 (CLSI) [15]. Briefly, C. albicans ATCC 10231 was inoculated on potato dextrose agar plate and cultured for 24 hours at 35 °C. Five colonies of about 1 mm in diameter were picked and suspended in 5 mL of physiological saline. The suspension was then adjusted to approximately 10⁶ CFU/mL with RPMI 1640. For the antibacterial assay, S. aureus ATCC 25923 was inoculated on a Mueller–Hilton broth agar plate and cultured for 24 hours at 37 °C. Five colonies of about 1 mm in diameter were then picked and suspended in 5 mL of physiological saline. The suspension was then adjusted to approximately 10⁶ CFU/mL with Mueller–Hinton broth and 2 µL of 2-fold serial dilution of each compound (in DMSO) was added to each row in the 96-well microplate, which contained 78 µL of microbe suspension in each well. Amphotericin B and vancomycin were used as positive controls for fungi and bacteria, respectively; DMSO was used as negative control. The 96-well plates were incubated at 35 °C aerobically for 24 hours. The MIC was defined as the minimum concentration of the compound that prevented visible growth of the microbes.

### 4. Conclusions

In summary, nine new secondary metabolites, including six oospolactone analogues, 7-hydroxyoospolactone (1), 7-methoxyoospolactone (2), 7-methoxy-9-hydroxyoospolactone (3), 10-acetoxy-9-hydroxyoospolactone (4), 6-dehydroxyssescandelin (5), and parapholactone (6), three compounds with a rare skeleton of isocoumarin coupled with phenylethylamine, namely paraphamide A (12), paraphamide B (13), and paraphamide C (14), to-
together with five known compounds, oospolactone (7), 8-O-methoxyoospolactone (8), 10-hydroxyoospolactone (9), 9,10-dihydroxyoospolactone (10), and oospoglycol (11), were isolated from the marine-derived fungus Paraphoma sp. CUGBMF180003. The carbon skeleton of 12–14 was identified as the first example of a coupled structure of isocoumarin and phenylethylamine. Natural isocoumarin analogues display a variety of bioactivities, including inhibitory activity against α-glucosidase [16], cytotoxicity [17,18], antifungal [19] and antibacterial activities [20–24], and anti-influenza virus [25]. The new isocoumarin analogues 1 and 6 showed inhibitory activity against S. aureus with MIC values of 12.5 μg/mL, but were devoid of growth-inhibitory activity against C. albicans activity at a concentration of 200 μg/mL.

Supplementary Materials: The following are available online at https://www.mdpi.com/article/10.3390/md19060313/s1, Figures S1–S53: HRESIMS, 1D and 2D NMR for compounds 1–6 and 12–14; Figure S54: Phylogenetic tree of strain CUGBMF180003.

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