For communities in developing countries, the majority of drinking water-related issues are due to pathogens from poor sanitation, resulting in infection and diarrhea. One cause of this is that these communities often do not have access to centralized water treatment facilities. Point-of-use (POU) systems are key solutions for treating water in developing communities; they are typically user-friendly, low cost, low maintenance, and grid-independent. Importantly, they treat and reduce the number of pathogens in water supplies, and many POU systems have been deployed and used by these communities, improving their livelihood. This review focuses on POU systems that cater to households or communities, with the aim to examine and evaluate technologies that have been implemented in POU systems in the past decade.

**TECHNOLOGIES FOR POU WATER TREATMENT**

Although there are many advanced water treatment technologies available, the technologies implemented in low-cost POU treatment systems are mainly characterized as mature and fundamental, aiming to remove waterborne pathogens. The available technologies are mainly flocculation and coagulation, filtration, and disinfection. Flocculation and coagulation remove the turbidity in the water, which reduces the supporting structure of microorganisms. This resulted in the removal of microorganisms. Filtration removes microorganisms by size exclusion, whereby microorganisms larger than the pore size of the filter will be retained within the system. Disinfection is the inactivation and
Figure 1: Photographs of (left) bioflocculant and (right) WaterROAM's portable membrane system (reproduced with permission from WaterROAM).

destruction of microorganisms in the system to a safe level.\textsuperscript{14} Photographs of a flocculant and a membrane system are shown in Fig. 1.

Coagulation and flocculation

Coagulation and flocculation are low energy and reliable water treatment processes. Coagulation is the destabilization of smaller particles (0.01–1 µm), resulting in the formation of larger particles. Flocculation is the formation of flocs from the destabilized particles.\textsuperscript{15} The flocculated particles can then be removed via settling or filtration.

Coagulation is able to effectively remove turbidity, colloids,\textsuperscript{16} and protozoan cysts.\textsuperscript{11} Coagulation, however, is unable to achieve a stable coliform removal.\textsuperscript{17} It has to be coupled with additional treatment processes or can improve the overall pathogen removal efficiency. When coagulation is used as a pretreatment or together with a membrane process, it can improve the removal of Cryptosporidium,\textsuperscript{18} E. coli, and MS2.\textsuperscript{19} In addition, coagulation will improve the disinfection process by removing turbidity and reduce the scavenging of disinfectant by natural organic matters.\textsuperscript{20}

The effective use of conventional coagulants such as alum, iron salt, and lime requires proper training and technical skills. Therefore, in the past, it had been mostly deployed for communal use in rural regions. Direct household usage of coagulant is rare.\textsuperscript{16,21} However, in the past decade, optimization of coagulant dosage, development of natural coagulants, and addition of disinfectant have allowed manufacturers, such as Procter & Gamble Co. (P&G) and Poly Glu International Co. to produce coagulants/flocculants that are stored in small POU sachets. This makes it easier for usage and distribution.

The PUR sachet from P&G is a combined flocculation/disinfection system. Ferric sulfate flocculates the destabilized particles within the water, while the calcium hypochlorite disinfects the water.\textsuperscript{22} In a study in western Kenya, sachets from P&G of different dosages were used to treat turbid water of various sources. In the study, high dose formulation of the flocculant-disinfectant product achieved water quality with mean E. coli concentration of 0 mg/L. The low dose formula, when used in high turbid water, was unable to consistently achieve drinking water quality.\textsuperscript{17} The use of P&G's flocculant-disinfectant sachet in rural Guatemala reduced the episode of diarrhea per person by 23.8%.\textsuperscript{22} In a case study by P&G in Vietnam, the PUR sachets achieved >5 log virus removal. 4 log removal and 3.6 log removal were also achieved for Cryptosporidium and Giardia, respectively.\textsuperscript{22} The PUR sachet was also shown to be able to achieve >99% arsenic removal and >8.2 log removal of E. coli.\textsuperscript{21}

Polyglutamic acid (PGA) is a natural, biodegradable, and edible bio-flocculant that is used at industrial scale\textsuperscript{23} and is the main ingredient in the PolyGlu sachet. It is produced from Bacillus licheniformis and Bacillus subtilis.\textsuperscript{24} PGA has high flocculating activity and high yield, resulting in the production of a smaller volume of sludge. It has better performance than aluminum sulfate in turbidity removal.\textsuperscript{20} Unlike the inorganic coagulant/flocculant, suspended solids removal by PGA is dose-dependent and is independent of temperature and pH, making it an ideal option for treating a wide range of water.\textsuperscript{20,25} PGA is also capable of adsorbing copper ions, removing harmful copper ions from the water source.\textsuperscript{21} In addition, in a patent submitted by Masamichi Mutou, a mixture containing γ-PGA with cross-link product of γ-PGA, sodium, calcium, and aluminum was able to inactivate microorganism.\textsuperscript{26} The components mentioned in the patent are used in Pga21, an ingredient used in PolyGlu; hence reducing the need for additional disinfectant within the PolyGlu sachet. However, the loss of residual chlorine may result in recontamination.

Coagulation and flocculation have shown success in treating drinking water and are mainly used for high turbid water. However, coagulation and flocculation cannot be a standalone water treatment solution. In the PUR sachet, disinfectants were added for effective removal of the pathogen. PolyGlu, on the other hand, requires proper water storage to prevent recontamination. While the PUR sachet and PolyGlu sachet had shown positive results in the removal of arsenic and copper, more studies are required to evaluate the effectiveness in removal of other potential heavy metals that can potentially reside in water sources. In addition, POU coagulant and flocculant are for single-use. This makes the process expensive in the long run. Ferric sulfate, a common coagulant, has shown potential in recovery and reuse.\textsuperscript{22} It can be explored to lower the operating expense (OPEX) of drinking water treatment in developing communities. Electrocoagulation, a process that generates aluminum hydroxide coagulant, has also shown potential in decentralized water treatment. Wind-powered electrocoagulation system was capable of removing up to 90% of microalgae and 97% of dissolved dye from synthetic wastewater within 72 h.\textsuperscript{34} This process demonstrated its ability to remove both pathogen and dissolved chemicals.

Filtration

Filtration is a simple water treatment process capable of removing colloids, suspended solids, and pathogens from drinking water sources. Its main removal mechanism is size exclusion. A well-designed filtration system will be able to generate a clean stream of drinking water. The size of different constituents commonly found in water sources and the pore size of conventional filtration process are summarized in Fig. 2. Filtration technologies implemented in POU are mainly biosand filtration (BSF) and membrane filtration.

In membrane filtration, external driving force is typically required to achieve the desired flow rate, and the external driving force required shall depend on the membrane pore size, surface area, and influent water quality. Compared to a membrane with larger surface area and pore size, higher applied pressure is required to filter water across a smaller membrane area with smaller pore size. In addition, maintenance is required in the filtration process. After prolong period of filtering, a foulant layer will be deposited on the filter layer. In sand filtration, flux can be recovered by backwash. In membrane system, backwash removes the reversible fouling, while chemical cleaning is required to remove biofouling and scaling.
**Biosand filtration.** BSF is a promising POU treatment process with more than 500,000 people using it worldwide. It can be easily constructed using raw materials that are locally sourced. The concept of BSF is similar to that of a conventional slow sand filter. However, BSF experiences varying flowrate and intermittent filtration through the sand layer. The outlet of the POU is located higher than the sand layer, allowing the sand layer to be saturated with water throughout the operation. This allows for microbial growth, developing biofilm around the filter media and sand particles. Excessive biofilm growth in the biolayer forms the Schmutzdecke, which removes larger microorganisms, colloids, and contaminants from the water sources.²⁵,³⁶

BSF is capable of reducing turbidity and removing *E. coli*. While a new BSF POU filter only achieved 63% *E. coli* removal, 98% removal was observed in a mature filter.³⁵ However, the treated water does not meet the WHO drinking water guideline. By replacing the sand with iron-coated sand, more than 2 log removals of *E. coli* was observed when it was fed with water with > 10⁹ CFU/mL *E. coli* concentration.³⁷ Increasing the flowrate, however, reduced the retention time and overall effectiveness of the filter.³⁸ This indicated that the sizing of the BSF is crucial in order to accommodate the target loading water volume. Removal of coliform will not be effective if the system is under-designed. Virus removal was also observed in BSF, achieving 1.3 log removals of echovirus 12 with fresh BSF and 2.2 log removal after maturation.³⁹ Filter depth was found to affect the removal of virus. A BSF with 5.4 cm filter depth achieved 6.7 log reduction of MS2, while a BSF with 5.4 cm filter depth achieved 3.4 log reduction of MS2.³⁶ In a separate study, BSF with 10% zero-valent iron achieved more than 5 log MS2 and rotavirus removal.³⁹ This indicated that virus removal was mainly due to sorption to the filter media.

Anaerobic BSF, with addition of ferrous iron, was also found to be able to reduce arsenic contamination in water.⁴⁰ These studies showed that the filter media would affect the type of contaminant removed by BSF and the potential to remove other contaminants.

BSF has shown success in reducing diarrhea episode globally. It has also been shown to have a long filter life, with a BSF in Cambodia lasting up to 8 years of continuous usage.⁴¹ BSFs have been deployed in real-world condition to treat water sources. The performance of BSFs deployment in various countries is summarized in Table 1.

**Membrane filtration.** Membrane filtration is a mature technology and is one of the most effective drinking water treatment processes. Clasen et al.⁴² concluded that membrane filtration is the most effective means of preventing diarrhea. It provides an absolute barrier for microorganisms, retaining them within the water source. However, membrane filters require an external

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### Table 1. Case studies of BSF deployments

| Location                  | Setting          | Study period | No. of household | *E. coli* reduction | Diarrhea episode reduction/age group | Reference |
|---------------------------|------------------|--------------|------------------|---------------------|--------------------------------------|-----------|
| Copan, Honduras           | Rural            | 11 months    | 195              | 53–69%              | 45% (<5 years) 39% (all ages)        | Aceituno et al.⁹¹ |
| Tamale, Ghana             | Rural            | 7 months     | 260              | 85–97%              | 74% (<5 years) 60% (all ages)        | Stauber et al.⁹² |
| Cambodia                  | Rural            | 5 months     | 336              | 60–89%              | 44% (2–4 years) 44% (all ages)       | Liang et al.⁹¹ |
| Bona, Dominican Republic  | Semi-urban/urban | 6 months     | 154              | 83%                 | 72% (2–4 years) 60% (all ages)       | Stauber et al.⁹³ |
| Kenya                     | Rural            | 9 months     | 59               | 94.4% (fecal coliform) | 51% (<5 years) 54% (all ages) | Tiwari et al.⁹⁴ |
| Haiti                     | Rural            | N.A.         | 107              | 98.5%               | N.A.                                 | Duke et al.⁹⁵ |

*Low *E. coli* reduction was mainly due to recontamination in storage tank.

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**Fig. 2** Size of water constituents and pore size of various filtration technologies (adapted with permission from ref.⁵² copyright Elsevier 2011)
of pathogen removal is dependent on the biochar characteristics and operating condition. While BSF is able to reduce pathogens in the water, the absence of residual disinfectants could allow recontamination to occur. In addition, disinfection process is required downstream to ensure the treated water is capable of meeting the WHO drinking water guidelines.

While both BSF and membrane POU systems are able to remove pathogens effectively, they are unable to remove dissolved solids. The pore sizes of the MF or UF membranes are too large to retain dissolved solids, allowing the dissolved solids to permeate through. Dissolved solids such as heavy metals can be removed by dense membrane systems such as the reverse osmosis (RO) process, which is a commonly used dense membrane to remove dissolved solids from water sources in desalination. However, due to the complexity of RO system, requirement for stable electricity supply, and maintenance, RO systems are not commonly deployed in rural areas. In addition, the use of RO also increases the operation cost of the water treatment system. Another dense membrane process, forward osmosis, is a favorable technology in treating chemically contaminated wastewater. Products from Hydration Technology Innovations had developed POU systems that utilized forward osmosis to filter water source. The draw solution, however, is not reusable and the system has a relatively low water flux.

Unfavorable water condition may result in rapid fouling of membrane, increasing maintenance frequency, and reducing water production. Water with elevated concentration of dissolved organic matters (DOM) will result in organic fouling. While both BSF and membrane POU systems are able to remove DOM, the accumulation of particles on membrane surface as well as within the membrane pores results in particle fouling. This fouling will reduce permeate flux and water filtration rate. Chemical cleaning and backwash will be required once the flux is below acceptable limits. In cases where water conditions are unfavorable, pretreatment is required prior to membrane process.

### DISINFECTION

Unlike the abovementioned two technologies, where microorganisms are removed from the water, disinfection resulted in the inactivation or death of microorganisms. Due to different microorganisms that potentially exist in water source, the type of disinfectant and dosage/exposure time differ from one microorganism to another. In addition, different disinfectants have different microorganism inactivation/destruction mechanisms.

| Manufacturer | Product | Filter (pore size) | Mode | Flow rate | Unit cost | Pathogen (removal) |
|--------------|---------|-------------------|------|-----------|-----------|-------------------|
| Vestergaard (Switzerland) | Lifes straw Family | Hollow fiber (0.02 µm) | Gravity | N.A. | USD89.95* | E. coli (>7 log) |
| Vestergaard (Switzerland) | Lifes straw community | Hollow fiber (0.02 µm) | Gravity | N.A. | USD329.95* | Cryptosporidium oocysts (>3.9 log) |
| Hydrologic (Cambodia) | Super Tunsai water filter | Ceramic (0.2 µm) | Gravity | 2–3 L/h | USD38 | Bacteria (>4 log) |
| Wateroam (Singapore) | ROAMplus | Polyethersulfone (0.02 µm) | Integrated pump | 250 L/h | USD260 | Coliform (>7 log) |
| Villagepump (Netherlands) | Villagepump 500 | Ultrafiltration | Integrated pump | 500 L/h | N.A. | N.A. |
| Icon Lifesaver (UK) | Lifesaver Jerycan | Ultrafiltration | Integrated pump | 75 L/h | £203.99 | R. terrigena (>6 log) |
| Skyjuice foundation (Australia) | Skyhydrant | PVDF (0.04 µm) | Gravity | 1000 L/h | N.A. | MS2 virus (>3 log) |

*Price obtained from eartheasy.com
Solar disinfection (SoDis) is a low-cost effective disinfection POU that involves the prolong exposure of water in polyethylene terephthalate (PET) or glass bottle under sunlight, allowing UV to transmit into the water. Prolong exposure also increases the temperature of the water in the bottle. Pathogens are inactivated by both the UV from the sun and the increase in water temperature.

There were many research on SoDis in the past 2 decades. Berney et al. revealed that inactivation of *Vibrio cholerae* starts once water temperature is risen beyond 40 °C. *E. coli* were found to have slightly higher heat resistant. Reduction in *E. coli* occurred when water temperature was risen beyond 45 °C. Temperature itself, however, is insufficient in disinfecting water source. In a separate study, a transparent bottle performed slightly better than a fully black bottle (negating effect of UV), even though the temperature within the transparent bottle was lower. After 4 h of solar exposure, coliforms were not detected in the transparent bottle but were detected in the fully black bottle. SoDis is also effective in turbid water. Though turbidity reduced the effectiveness of transmittance of UV, the scattering of UV light within the bottle improved the efficiency. A study revealed that SoDis was able to disinfect and achieve 6-log *E. coli* removal with well water of turbidity up to 100 NTU.

Research had been done to improve the performance of SoDis. The design of the reflector and bottle can affect the SoDis treatment. Kalt et al. had demonstrated that a double parabolic collector was able to increase the water temperature to above 45 °C within 42 min, with *E. coli* concentration below detection limit after 20 min. In another study, transparent bottles on a double parabolic concentrator and large concentrator were able to maintain temperature over 45 °C for 3 h and 6 h, respectively. Total *coli*form concentration of both concentrator systems was below the detection limit within 4 h. In the same study, the use of a half-black bottle on the large concentrator was able to maintain temperature over 45 °C throughout the day, with peak temperature reaching 75 °C. Bottles and bags made from other materials can also be used for SoDis. Polyethylene (PE) bag and polypropylene copolymer (PPCO) outperformed PET bottles due to its higher UVB transmittance, with more than 2 log bacteria removal at 450 kJ/m² of UV dose. However, gas chromatography–mass spectrometry of water samples revealed degradation of low density PE bag after prolong solar exposure, producing 2,4-di-tet-butylphenol. Addition of addictive (100 mg/L sodium percarbonate, 100 mg/L ascorbic acid, and 20 µg/L copper sulfate) improved inactivation rate of *E. coli*, reducing the 3 log inactivation time from 2.93 to 1.95 h.

SoDis has proven to be a simple, low-cost, and effective disinfecting technique. It has been deployed in different places around the world, serving more than 30 countries worldwide. Field studies of SoDis are summarized in Table 3. However, SoDis does have its disadvantages; it is heavily dependent on weather condition; and it requires an exposure period of more than 48 h on cloudy days. Commercial product like WADI can assist users with a more accurate determination of whether the water had been disinfected. In addition, SoDis is unable to remove chemical pollutants in the water.

### Nanotechnology

Extensive research in nanotechnologies in the past 2 decades have opened new door for development of water treatment techniques. Nanoparticles, with sizes between 1 and 100 nm, exhibit different characteristics and have a higher specific surface area than their conventional macro-scale counterparts. This makes nanoparticles more efficient in water treatment. Nanoparticles have been applied in various fields such as adsorption, membrane development, and advanced oxidation processes. For instance, development of Fe-impregnated carbon nanofiber has shown potential in arsenic removal via adsorption, while nanomaterials added new functionalities and benefits into membrane. Nanotechnology had been applied in POU as well, namely, TiO₂ and Ag nanoparticles.

*Titanium dioxide nanoparticles.* Titanium dioxide is a widely used photocatalyst; it is applied in many products and is the most produced nanoparticle. Upon exposure to UV–vis light, TiO₂ generates OH radicals, H₂O₂, O₂⁻, and O₂, which can oxidize organics and inactive microorganisms. During the process, TiO₂ is regenerated, making it available for further reactive oxygen species generation. This photocatalytic property allows TiO₂ to be applied in different water treatment processes. TiO₂ nanoparticles have been applied onto the membrane, reducing biofouling and increasing hydrophilicity. In the past decade, titanium dioxide has made its way into commercial POU water treatment. Solarbag, a product by Puralytics, lined TiO₂ nanoparticles within the bag to remove contaminants and disinfect the water.

Research has shown that TiO₂ can enhance SoDis effect. TiO₂ immobilized within the water treatment system was able to reduce *E. coli* concentration to below detection limit within 4 h, while an uncoated system took more than 5 h. It has also been reported that suspended TiO₂ yield better results than immobilized form due to the increased surface area. However, retaining the suspended TiO₂ and separating them from the treated water proved to be a challenge. A review by Laxma Reddy et al. summarized the photocatalytic disinfection property of TiO₂ on bacteria, fungi, virus, protozoa, and algae.

*Silver nanoparticles.* Silver nanoparticles (AgNP) is another nanotechnology that is widely used in consumer product. Though its production is 2% of TiO₂ nanoparticles, it has been applied in over 400 commercial products due to its antimicrobial properties. It is a well-established fact that silver is toxic to microorganisms. However, the bactericidal effect of AgNP differs from silver ions. Silver nanoparticles attached to the bacterial cell membrane electrostatically, disrupting its permeability and damaging it.

### Table 3. Field studies of SoDis

| Location   | Setting | Period | Study group | Diarrhea episode reduction/age group | Reference          |
|------------|---------|--------|-------------|--------------------------------------|--------------------|
| Kenya      | Rural   | 12 weeks | 206 children | 9% (5–16 years old)                  | Conroy et al.³⁷     |
| India      | Urban   | 6 months | 200 children | 41% (≤5 years)                       | Rose et al.⁹⁷       |
| Bolivia    | Rural   | 1 year  | 376 children | 33% (≤5 years)                       | Mausezahl et al.⁶⁴ |
| Cambodia   | Rural   | 1 year  | 928 children | 50% (≤5 years) 63% (≤5 years)         | McGuigan et al.⁹⁸  |

*Dysentery diarrhea

*Non-dysentery diarrhea*
can also enter and bioaccumulate within the bacterial cell and interact with its DNA, damaging the cell.\textsuperscript{76,77} Inhibition of bacteria growth was observed at 5 mg/L of AgNP,\textsuperscript{78} with 100% inhibition at 50 mg/L of AgNP.\textsuperscript{75} Ceramic filter coated with AgNP was also effective in bacteria removal, with 92.82\% \textit{E. coli} removal and 87.11\% coliform removal.\textsuperscript{79} By coating AgNP onto polyurethane filters, more than 5 log bacteria removal was observed.\textsuperscript{80} AgNP is also toxic to virus such as norovirus\textsuperscript{81} and poliovirus.\textsuperscript{82} However, AgNP was found to be ineffective against MS2 bacteriophage.\textsuperscript{83}

Silver nanoparticles has made its way into POU water treatment systems for the developing communities. Tata Swach, a water filter that uses rice husk ash with AgNP, is able to achieve 1 log \textit{E. coli} removal.\textsuperscript{84} Folia filter by Folia water is another filter coated with copper and silver nanoparticle for disinfection purpose. A research by the co-founder, Dankovich,\textsuperscript{85} showed that more than 6 log \textit{E. coli} removal was achieved with 1.6 mg AgNP/g filter paper. Field study revealed more than 4 log \textit{E. coli} removal by copper and silver nanoparticles coated filter paper.\textsuperscript{86}

Drawbacks. Both TiO\textsubscript{2} and silver nanoparticles had showed great potential in advanced oxidation process and disinfection. However, the long-term effect of nanoparticles is largely unknown.\textsuperscript{87} Improper disposal and leaching can result in nanoparticles entering the environment. Nanoparticles can also enter aquatic animal by ingestion, gills, or body walls.\textsuperscript{88} TiO\textsubscript{2} and AgNP, at elevated concentration, are toxic to marine life;\textsuperscript{89} high concentration of AgNP can affect mammalian cells such as reducing cell vitality and mitochondrial function, and can cause cell membrane leakages.\textsuperscript{90} Therefore, proper disposal method should be incorporated into the systems, reducing the release of nanoparticles into the environment. In addition, although there are no guidelines on nanoparticles concentration in drinking water, leaching of nanoparticles into treated drinking water should be systematically investigated. Lastly, in region where nanoparticle technologies had been implemented, nanoparticle concentration can be monitored to observe the possible environmental fate and effects on the environment.

CONCLUDING REMARKS AND PERSPECTIVE

Technologies implemented in POU systems are effective in removing pathogens and have successfully reduced diarrhea episodes. The systems are simple to use with minimal maintenance. Optimization and development allowed flocculants to be packaged into POU sachets. Research in BSF has optimized the BSF depth and filtration flow rate. Development in membrane technologies has allowed water filtration process to be grid-independent. In addition, nanotechnologies have opened new opportunities for development in POU systems.

However, to successfully ensure removal of pathogens, multi-barrier approach should be adopted. Users can couple filtration with disinfection, such as BSF together with SoDis, to minimize the risk of bacterial infection. In addition, the use of external storage might result in recontamination, as shown in the case studies in BSF deployment. Users should have a clean bucket that is primarily used for drinking water storage.

Several challenges still need to be addressed for the above-mentioned technologies. Firstly, inorganic coagulants and flocculants have shown to be recoverable and reusable in drinking water treatment, reducing the OPEX. The recovery process, however, uses high concentration of alkaline and acid, resulting in high cost. Therefore, the stability of the recovered coagulant and the number of times it can be reused should be further investigated and optimized. Secondly, anaerobic BSF with ferrous iron has shown to be able to remove arsenic. This shed new light in the heavy metal removal by BSF technology. Arsenic poisoning is a serious problem encountered across the world. An improved arsenic removal could be achieved by varying the ferrous iron content, acetic acid addition, or the operating parameter of the BSF system. However, the mechanism of arsenic removal by the BSF system is still unknown. Nonetheless, this discovery could potentially widen its application in the treatment of arsenic-rich groundwater. In addition, biochar can be incorporated into BSF to improve pathogen and contaminant removal. Even though biochar needs to be replaced periodically, it can be made using locally sourced materials. The long-term effect of nanoparticles on the environment and human health in the rural setting has not been extensively monitored and evaluated. Improper disposal might release the nanoparticles into the environment. Suppliers and distributors should educate the users on proper disposal of nanoparticle products. In addition, the environmental fate of nanoparticles in the developing communities should be studied.

Lastly, while filtration and disinfection technologies are capable of removing pathogens from water sources, they are unable to effectively remove dissolved contaminants such as arsenic and chromium. These dissolved contaminants will cause chronic health issues if this water is consumed. Prior to consumption, pretreatment or post-treatment such as adsorption, coagulation, and iron exchange are required to reduce and remove these harmful contaminants if they are present.

AUTHOR CONTRIBUTIONS

C.K.P. conducted the detailed search and review of published relevant literatures and wrote the draft manuscript. H.Y.N. initiated this topic as a review paper, provided guidance on the content and structure of the manuscript, and amended the manuscript.

ADDITIONAL INFORMATION

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