Imparting biomolecules to a metal-organic framework material by controlled DNA tetrahedron encapsulation

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Recently, the incorporation of biomolecules in Metal-organic frameworks (MOFs) attracts many attentions because of controlling the functions, properties and stability of trapped molecules. Although there are few reports on protein/MOFs composites and their applications, none of DNA/MOFs composite is reported, as far as we know. Here, we report a new composite material which is self-assembled from 3D DNA (guest) and pre-synthesized MOFs (host) by electrostatic interactions and hydrophilic interactions in a well-dispersed fashion. Its biophysical characterization is well analyzed by fluorescence spectroscopy, quartz crystal microbalance (QCM) and transmission electron microscopy (TEM). This new composite material keeps 3D DNA nanostructure more stable than only 3D DNA nanostructure in DI water at room temperature, and stores amounts of genetic information. It will make DNA as a guest for MOFs and MOFs become a new platform for the development of DNA nanotechnology.

Metal-organic frameworks (MOFs) are permanently microporous materials synthesized by assembling metal ions with organic ligands in appropriate solvents1–4. They have crystalline structures and typically are characterized by large internal surface areas, uniform but tunable cavities and tailorable chemistry5,6. Their porous crystals are useful because they allow access of molecules through their pore apertures for gas storage, chemical separation7–9, catalysis10,11, drug delivery12–15 and sensing16–18. The pore apertures dictate the size of the molecules that may enter the pores, which provide the surface and space to carry out these functions19–22. By serving as unique host matrices for various functional species, they also offer the opportunity to develop new types of composite materials that display enhanced (gas storage) or new (catalytic, optical and electrically conductive) behaviors in comparison to the parent MOFs counterparts2,23.

In particular, the incorporation of biomolecules in MOFs attracts much attention because of controlling the functions, properties and stability of the trapped molecules. Biomolecule/MOFs composites can be prepared either by using MOFs as templates to generate biomolecules within their cavities or by encapsulating pre-synthesized biomolecules in MOFs. In the former case, small and naked biomolecules are generated and embedded in the cavities of MOFs. In the latter, however, usually the biomolecules are stabilized with certain surfactants, capping agents or even ions, and the biomolecules hydrodynamic radius are much larger than the cavity size of MOFs. The biomolecules do not occupy the MOFs cavities, but instead are surrounded by grown MOFs materials. For example, Fujita et al. utilizes self-assembly to construct sphere frameworks around a covalently tethered protein24. The protein was attached to one bidentate ligand and, upon addition of Pd (II) ions (M) and additional ligands (L), M12L24 sphere framework self-assembled around the protein. Omar M. Yaghi et al. reports a strategy to expand the pore aperture of MOFs into a previously unattained size regime25. The pore apertures of IRMOF-74-VII and IRMOF-74-IX are large enough for natural proteins, such as vitamin B12, myoglobin, and green fluorescent protein (GFP) to enter the pores respectively.

Although there are few reports on protein/MOF composites and their applications (mainly focused on controlling the proteins’ structure, making the functional proteins or enhancing protein stability), there are no reported examples of DNA/MOFs composites, as far as we know. This emerging area is characterized by
significant challenges, several of which we address here. First, effective control over the spatial distribution of DNA nanostructure within the MOFs matrix is largely lacking with existing strategies for encapsulation. In addition, the choice and synthesis of a suitable MOFs for DNA encapsulation is a daunting synthetic challenge due to size of DNA, especially 3D DNA nanostructure and the harsh environment of DNA stable existence. Third, the encapsulation of biomolecules will have a significant impact on their chemical properties and should lead to new application, including catalysis, sensing and keeping protein stable. In this regard, the DNA/MOFs composite material seems attractive, but as yet is not developed.

Due to the challenges and potential applications, there have been no reported examples of two dimensional (2D) DNA nanostructures encapsulation by artificial well-defined MOFs, let alone three dimensional (3D) DNA nanostructures. Here, we report a new composite material which is self-assembled from 3D DNA (guest) and pre-synthesized MOFs (host) in a well-dispersed fashion by electrostatic interactions and hydrophilic interactions. Its biophysical characterization could be well analyzed by fluorescence spectroscopy, quartz crystal microbalance (QCM) and transmission electron microscopy (TEM). We expect that 3D DNA encapsulation into the MOFs will allow us to explore the function of synthetic hosts for the conformational and functional control of encapsulated 3D DNA. And this new composite material could keep 3D DNA nanostructure more stable than only 3D DNA in DI water at room temperature and store a large amount of genetic information.

Previous studies have proven that 3D DNA nanostructure has excellent mechanical rigidity and structural stability. In our work, we choose the DNA tetrahedral nanostructure originally developed by Turberfield et al. as the guest. This tetrahedral nanostructure can be rapidly and reliably assembled by four designed DNA oligonucleotides and easily functionalized with different chemical moieties and biomolecules. Each edge of a tetrahedron is an 18-base-pair double helix, which makes the tetrahedral edges is 5.8 nm in length. To meet the require of large size and sensitive nature of DNA, we choose water-insoluble M12L24 sphere framework is self-assembled from 12 Pd (II) ions (M) and 24 bent ligands (L) as our first host. We dissolve it in 5% DMSO aqueous solution and detect the interaction between M12L24 sphere framework and 3D DNA by QCM. We find that the frequency signal change is due to the effect of DMSO, not because of M12L24 effect (Figure S7). Thus, M6L4 sphere framework is chosen due to its increased solubility in the aqueous solution mixture. It is a 12+ charged M6L4 type sphere framework with a diameter of 7 nm in length, constructed from ten species: 4 bent ligands (L) held together by 6 Pd (II) ions (M). Its tetrahedral symmetry provides a large central void for accommodating 3D DNA due to electron donation to Pd2+ centers at three pyridine coordination sites which acts as a strong electron acceptor. M6L4 sphere framework is prepared according to the method reported by Fujita et al. Generally, there are three assays to encapsulate DNA tetrahedron into the metal-organic framework. The first assay, aqueous solutions of M6L4 sphere frameworks (300 μM, 10 μL), four oligonucleotides (10 μM, 5 μL) and TM buffer (20 μL) are mixed briefly, heated to 95 °C for 2 min and immediately cooled to 4 °C in about 50 s using a PCR machine and then kept at room temperature for 48 h without stirring. But the reaction forms many precipitates during the self-assemble process of M6L4 sphere frameworks and DNA tetrahedrons, which indicates that the M6L4 sphere frameworks structure are inevitably collapsed at 95 °C in TM buffer, which indicates the assay is not successful to fabricate composite materials. The second assay, the four stands of DNA oligonucleotides, organic ligands and metallic compounds are mixed in TM buffer, but the organic ligands cannot dissolve in DI water which indicates that this method is negative to the forming of M6L4 sphere framework. The third assay, we respectively synthesize 3D DNA and M6L4 sphere framework in aqueous solution, and then mix the two kinds of aqueous solution directly. Keep them at room temperature for 48 h without stirring. Finally, we obtain the transparent liquid. So in this work, we choose the third method to get the new composite material.

To investigate whether 3D DNA could bind with the M6L4 sphere framework, we use FRET measurements by using fluorescence spectrophotometer (Figure 1a). It can confirm the existence of a fluorescence resonance energy transfer pair from donor to acceptor. We encapsulate bisanthracene into M6L4 sphere framework to form a new photoactive exciplex (the FRET donor). The DNA tetrahedron is labeled with FAM on an un-hybridized "hinge" base (FL-3D DNA, the FRET acceptor). If they can be attracted by electrostatic interactions, forming a new composite material, the donor-acceptor distance would be less than 10 nm, corresponding FRET efficiencies of approximately 90%. In contrast, if they don't have any interaction, the donor–acceptor distance would be greater than 10 nm, corresponding FRET of less than 5% as shown in Figure 1b, there is a weak peak at 518 nm of the FL-3D DNA. While, mixing the FL-3D DNA and the photoactive exciplex together, there is a new strong emission at 518 nm. It demonstrates that the donor-acceptor distance less than 10 nm. The FRET measurements show that not only electrostatic attractions but also hydrophilic interactions play important roles in the self-assembling process of FL-3D DNA and M6L4 sphere framework.
To test whether the 3D DNA and the M₆L₄ sphere framework can self-assemble to form a new composite material, we design the 3D DNA structure with three thiol groups at three vertices (HS-3D DNA). It can be self-assembled onto Au surfaces by their thiol groups, as monitored in real time by a surface-sensitive acoustic method, quartz crystal microbalance (QCM) (Figure 2a). According to Sauerbrey et al. reports, the change in resonance frequency Δf is proportional to the change of the adsorbed mass Δm. Hence, any adsorption/desorption process at the quartz interface can be monitored. A typical example of a time resolved QCM measurement is shown in Figure 2b. The decrease of frequency is about 100 Hz in 15 minutes, which indicates HS-3D DNA can be anchored at Au surfaces by Au-S chemical bond within 15 minutes, corresponding to the “rapid” phase (Figure 2b red space). Meanwhile, the remaining self-assembled process from B to C takes 3 hours, corresponding to the “slow” process, and the change of frequency is less than 20 Hz in 3 hours, much slower than the first 15 minutes (Figure 2b blue space). When washed by TM buffer (from C point), the frequency keeps almost constant, which indicates that nearly all the HS-3D DNA are assembled onto the Au surface. However, in the presence of M₆L₄ sphere frameworks (from D point), a decrease in resonance frequency is observed, suggesting the adsorption of M₆L₄ sphere frameworks (Figure 2c red space). But addition of more M₆L₄ sphere frameworks doesn’t alter the QCM response (from E point to F point). It is mainly due to that the self-assemble process of HS-3D DNA and M₆L₄ sphere frameworks are saturated. Then the system is flushed with TM buffer (from F point). The frequency does not change, which shows that the HS-3D DNA and the M₆L₄ sphere frameworks have strong binding interactions. The QCM results demonstrate that HS-3D DNA and the M₆L₄ sphere framework can self-assemble to form a new stable composite material by electrostatic attractions and hydrophilic interactions.

FRET and QCM results reveal that the 3D DNA and M₆L₄ sphere frameworks are attracted by electrostatic attractions and hydrophilic interactions which can form a stable composite material. When we detect the interaction between M₁₂L₂₄ sphere framework and 3D DNA by QCM, the frequency signal change is due to the effect of DMSO, not because of M₁₂L₂₄ effect (Figure S7) that indirectly demonstrates that HS-3D DNA is encapsulated into M₆L₄ sphere framework (Figure S8). TEM is utilized to identify whether the DNA tetrahedrons are encapsulated into the M₆L₄ sphere frameworks (Figure 3a). Figure 3b shows TEM images of M₆L₄ sphere frameworks. Because the M₆L₄ sphere framework is translucent stone, light is allowed to penetrate the entire sample, enabling us to systematically section from the top to the bottom. M₆L₄ sphere frameworks are independent but not well distributed in DI water. We find different shapes of the M₆L₄ sphere frameworks according to the TEM images. Some of them are aggregated together, because of the different tilting angles when we take images. Due to the low contrast of DNA nanostructure in TEM test, we label the 5’ end of each strand to form a tetrahedron with AuNPs (5 nm in diameter). Figure 3c is the TEM images of AuNPs modified DNA tetrahedron (AuNPs-3D DNA). According to the images, we can confirm that it is truly the structure of the DNA tetrahedral. Figure 3d is the TEM images of the

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**Figure 2 | QCM detection.** (a) Schematic representation of the formation of the proposed composite material. Step 1, the HS-3D DNA is expected to readily anchor at the Au surface by Au-S chemical bond. Step 2, M₆L₄ sphere frameworks are expected to be attracted to the platform by electrostatic interaction and hydrophilic interaction. (b) The self-assemble process of HS-3D DNA at the Au surface is monitored in real-time using QCM. It can be anchored at Au surfaces by Au-S chemical bond within 15 minutes, corresponding to the “rapid” process (red space). Meanwhile, the remaining self-assembled process from B to C takes 3 hours, corresponding to the “slow” process, and the change of frequency is less than 20 Hz in 3 hours, much slower than the first 15 minutes, corresponding to the “slow” process (blue space). It indicates that most of the HS-3D DNA are assembled on the Au surface. (c) The self-assemble process of DNA tetrahedrons and M₆L₄ sphere frameworks is monitored in real-time using QCM. In the presence of M₆L₄ sphere frameworks, a decrease in resonance frequency is observed, suggesting the adsorption of M₆L₄ sphere frameworks (red space). Then the self-assemble process is saturated (from E point to F point). But when the system is flushed with TM buffer (from F point to the end), the frequency does not change, which shows that the HS-3D DNA and the M₆L₄ sphere frameworks have strong binding interactions. All the experiments are carried out at room temperature.
composite material, obtained by mixing AuNPs-3D DNA buffer solution and M₆L₄ sphere framework aqueous solution, keeping at room temperature for 48 h without stirring. TEM images reveal that multiple M₆L₄ sphere frameworks contain a AuNPs-3D DNA, yet well disperse. But according to the images of the composite compounds, we can confirm that it is truly the structure of the AuNPs-3D DNA. It is obtained by mixing AuNPs-3D DNA buffer solution and M₆L₄ sphere framework aqueous solution, keeping at room temperature for 48 h without stirring. TEM images reveal that multiple M₆L₄ sphere frameworks contain a AuNPs-3D DNA, yet well disperse. All the experiments are carried out at room temperature. 2D DNA nanostructure cannot be stable in DI water, let alone 3D DNA nanostructure. Given 3D DNA can be encapsulated into M₆L₄ sphere frameworks, this new composite material may be useful for keeping 3D DNA nanostructure stable. To illustrate this, we monitor the frequency signal changes of the composite material and 3D DNA when they are washed by DI water in real-time using QCM. HS-3D DNA is assembled from 3 thiolated DNA fragments and 1 none modified DNA fragment (All of the 4 strands contain 55-nt nucleotides). It can anchor at Au surface by Au-S chemical bond for the 3 thiolated DNA fragments. The HS-3D DNA nanostructure, however, will disassemble when it is washed by DI water due to the weak hydrogen bonding force. The 1 none modified DNA fragment is washed out of the Au surface. So there is about 25% (1 out of 4 strands) frequency increase. Just as Figure 4 shown, when HS-3D DNA which is layered at Au surface (Figure 4 red space, black line) is washed by DI water, the increase of frequency is about 30 Hz in 300 minutes (Figure 4 blue space). The frequency signal change is about 25% relative to 120 Hz (Figure 2b), which indicates that HS-3D DNA nanostructure is disassembled. However, addition of M₆L₄ sphere frameworks and forming a new composite material (Figure 4 red space) is washed by DI water, the frequency keeps almost constant (Figure 4 blue space, red line), which suggests that HS-3D DNA nanostructure is stable. According to the QCM data, we also believe that this new composite material (M₆L₄-3D DNA) can keep 3D DNA stable.

Figure 3 | TEM analysis. (a) Schematic illustration of the forming of the composite material for TEM testing. (b) TEM images of M₆L₄ sphere frameworks. Due to the M₆L₄ sphere frameworks are not well distributed in DI water and the different tilting angles when we take images. We find different shapes of the M₆L₄ sphere frameworks. (c) TEM images of AuNPs modified DNA tetrahedral (AuNPs-3D DNA). Because of the low contrast of DNA nanostructure in TEM test, we label the 5' end of each strand to form a tetrahedron with four AuNPs (5 nm in diameter). According to the images, we can confirm that it is truly the structure of the AuNPs-3D DNA. (d) TEM images of the new composite material. It is obtained by mixing AuNPs-3D DNA buffer solution and M₆L₄ sphere framework aqueous solution, keeping at room temperature for 48 h without stirring. TEM images reveal that multiple M₆L₄ sphere frameworks contain a AuNPs-3D DNA, yet well disperse.
nanostructure more stable than only 3D DNA nanostructure in DI water.

In summary, we successfully synthesize a new composite material which is self-assembled from 3D DNA (guest) and pre-synthesized MOFs (host) by electrostatic attraction and hydrophilic interaction in a well-dispersed fashion. This new composite material can keep 3D DNA nanostructure more stable in DI water than only 3D DNA nanostructure in DI water and store a large amount of genetic information. It will become a general platform in the development of structural DNA nanotechnology. So this composite material will make not only DNA as a guest for the host of MOFs, but also MOFs become a new platform for the development of DNA nanotechnology.

**Methods**

M₄L₄ sphere frameworkˢ, M₄L₄ sphere framework¹ and DNA tetrahedron² are synthesized according to previous protocol. A detailed description of the protocol can be found in the supporting information.

**FRET measurements.** A quantity of 100 μL of 1 μM FAM-labeled tetrahedron solution is incubated with 100 μL of 10 μM host-guest exciplex at room temperature for 24 h, and then the solution is directly used for fluorescence. The fluorescence emission spectrum from 410 to 600 nm is measured at 25°C in a quartz cell with 10 mm path length using 376 nm excitation wavelength.

**QCM detection.** The experimental setup for the QCM-measurements used in the present study is described in more detail elsewhere. Au chip used only for QCM detection is first loaded onto a QCM sensor, and then TM (pH 10 mm path length using 376 nm excitation wavelength.

**Image processing.** Mixed the four strands AuNPs modified ssDNA in equimolar quantities in 1× TBE with 50 mM NaCl, heated the mixture to 95°C for 2 min and immediately cooled to 4°C in about 50 s. Then the mixture incubate for 24 h at room temperature to give high yield Au modified DNA tetrahedron. 10 μL solution of the AuNPs Modified DNA tetrahedron is incubated at room temperature in the presence of 100 μL (300 mM) PD-cage for 1 day. Then the samples are transferred to a copper wire mesh, allowed to dry. Images are collected on a transmission electron microscope at an acceleration voltage of 200 kV.

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Author contributions
Y.M.J., B.M.W. and F.X. designed the study. Y.M.J. performed all chemical synthesis, conducted the FRET, TEM experiments, performed all data analysis and drew all schematic diagrams. B.M.W. conducted the QCM experiment and performed the data analysis. Y.M.J. and F.X. prepared the figures and co-wrote the paper. Y.M.J. and B.M.W. contributed equally to this work. Y.M.J., B.M.W., R.X.D., Y.Z., B.Y.W., A.H., N.N.L., X.W.O., S.F.X., Z.P.C., X.D.L. and F.X. reviewed the manuscript.

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