Effects of renewable fuel and exhaust aftertreatment on primary and secondary emissions from a modern heavy-duty diesel engine

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\textbf{ABSTRACT}

Compared to petroleum diesel, renewable diesel fuels and exhaust aftertreatment systems can reduce primary exhaust emissions that are hazardous to human health and the environment. Secondary aerosol emissions which form upon atmospheric processing have, however, been less studied. This study aimed to quantify the impacts of replacing petroleum diesel with renewable fuels (hydrotreated vegetable oil [HVO] and rapeseed methyl ester [RME]) on primary and secondary aerosol emissions from a heavy-duty diesel engine at different stages of an exhaust aftertreatment system. Emission characterization was obtained by combining a battery of physical characterization techniques with chemical characterization using aerosol mass spectrometry.

At engine-out measurements, RME and HVO reduced primary particulate matter (PM) emissions (for example equivalent black carbon [eBC]) and secondary aerosol production (studied with an oxidation flow reactor [OFR]) by mass compared to petroleum diesel. The diesel oxidation catalyst (DOC) reduced primary nucleation mode emissions, reduced the effective density of soot mode emissions, and reduced secondary particle production by mass. The DOC + a diesel particulate filter removed >99% of the particle number and eBC emissions.

Volatile PM emissions (for example organic aerosol) were found to be distributed between the nucleation mode and soot mode for both primary and secondary emissions, to a degree that depends on both fuel type and aftertreatment. A high mass concentration of condensable species and a low condensation sink in the soot mode led to increased fractions of condensable species present in the nucleation mode. Aging in the OFR led to increases in particle effective density. Motoring the engine (running without combustion) showed that the nucleation mode originated primarily from lubricating oil, and nonvolatile nanoparticle emissions were identified down to 1.2 nm in particle size.

In conclusion, replacing petroleum diesel with HVO and RME changes emission characteristics and can help reduce key aerosol emissions of relevance for adverse health and climate impact, especially for diesel engines with no or limited exhaust aftertreatment.

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1. Introduction

Vehicle emission legislation has become increasingly stringent in the past 20 years in order to combat air pollution and reduce carbon dioxide (CO$_2$) emissions. Emission legislation and sustainability goals have forced the development and implementation of various emission controls, such as more efficient engines with lower emissions, exhaust aftertreatment systems (EATs) and renewable diesel fuels to reduce the emissions. Primary emissions and the subsequent secondary organic aerosol (SOA) formation from combustion engines contribute considerably to urban and regional particulate air pollution (Gentner et al., 2017; Hallquist et al., 2009; Robinson et al., 2007). In this study, we investigate the effects of renewable diesel fuels and EATs on the primary and secondary aerosol emissions of a heavy-duty diesel engine.

Emissions of untreated diesel exhaust (engine-out) are composed of major gas-phase species such as CO$_2$, carbon monoxide (CO), nitrogen oxides (NO$_x$), a complex mixture of gaseous hydrocarbons (HC), and of particulate matter (Matti Marić, 2007). The particulate emissions from diesel engines are often characterized by a bimodal particle size distribution, with nucleation mode particles dominating the number concentration and soot mode particles dominating the total emitted mass (Burtscher, 2005; Matti Marić, 2007). The soot mode consists of agglomerates of insoluble primary soot particles, primarily made up of elemental carbon (EC) arranged in graphitic-like material, with a varying degree of adsorbed or condensed volatile organic and inorganic matter on the surfaces (Kittelson, 1998; Matti Marić, 2007). Early in the combustion process the soot matures to refractory black carbon (rBC). The concentration of rBC in the exhaust after the engine is often controlled by late cycle in-cylinder oxidation processes (Malmberg et al., 2017). Nucleation mode particles can be emitted as delayed primary emissions which form rapidly after cooling and dilution of the exhaust gas in the atmosphere in a sulfur-driven nucleation process (Karjalainen et al., 2014; Ronkkö et al., 2013). However, nanoparticles have also been found in primary emissions at high temperatures and may contain nonvolatile cores (nanoclusters) that can later act as additional condensation sinks for gas-phase species to condense on in the dilution process (Karjalainen, Ntziachristos, et al., 2016; Kuuluvainen et al., 2020). Karjalainen, Timonen, et al. (2016) found that the nucleation particles originated from the lubricating oil, and that the nonvolatile core was composed of heavy organic or organometallic material. A recent paper by Kuuluvainen et al. (2020) proposed that nucleation cores in heavy-duty non-road engine exhaust can be divided into two independent particle modes where one is lubricant originated and the other is fuel originated. However, only the emissions before any after-treatment (engine-out) were measured. When assessing atmospherically relevant emissions, the effect of EATs on the nucleation mode particles are of vital importance.

The composition of the particulate fraction of diesel exhaust depends strongly on the combustion conditions and the type of fuel (Burtscher, 2005). Oxygenated renewable fuels, such as fatty acid methyl ester (FAME) fuels, can reduce particulate matter (PM), HC, and CO emissions, but may simultaneously lead to an increase in NO$_x$ emissions (Knothe, Krab, & Van Gerpen, 2015; Mahmudul et al., 2017; Matti Marić, 2007). Exhaust PM from combustion with FAME fuels has been found to contain a higher organic carbon (Popovicheva et al., 2017) or soluble organic (Matti Marić, 2007; Murtonen, Aakko-Saksa, Kuronen, Mikkonen, & Lehtoranta, 2010) mass fraction compared to diesel. FAME fuels can also impact the soot characteristics, and smaller primary particle and agglomerate size have been observed (Savic et al., 2016). Paraffinic renewable diesel fuels, such as hydroprocessed vegetable oil (HVO), have in some studies been shown to reduce the PM, HC, CO, and NO$_x$ emissions (Dimitriadis, Natsios, Dimaratos, & Katsaounis, 2018; Gren et al., 2020; Kuronen, Mikkonen, Aakko, & Murtonen, 2007; Murtonen et al., 2010) while other studies have shown negligible effects in comparison to petroleum diesel (Mizushima, Kawano, Ishii, Takada, & Sato, 2014; Shukla et al., 2018). However, the PM reduction observed for combustion with HVO is generally smaller compared to the reduction observed with FAME fuels (Knothe et al., 2015; Na et al., 2015). The different characteristics of the exhaust from HVO and FAME fuels could as well have effects on the operation and efficiency of the EATs.

EATs are designed to reduce multiple pollutants found in diesel exhaust (Reşiţoğlu, Altinişik, & Keskin, 2015), and are often composed of a series of units that treat different combustion products in each step. A common approach is to utilize a combination of a diesel oxidation catalyst (DOC) to reduce gas phase HC and CO, a diesel particulate filter (DPF) to reduce PM, and a NO$_x$ reduction unit (Reşiţoğlu et al., 2015). The DOC also reduces the organic fraction of the exhaust PM (Liu, Eckerle, & Ottinger, 2018; Zeraati-Rezaei, Alam, Xu, Beddows, & Harrison, 2020). Internal combustion controls, such as exhaust gas recirculation (EGR), are an alternative to reduce NO$_x$ levels (Sorathia et al., 2012; Zheng, Reader, & Hawley, 2004), but at the expense of increased PM emissions (Gren et al., 2020). While the knowledge of primary emissions from renewable fuels is increasing, research is still needed on the impact of EATs in combination with renewable fuels and the subsequent aging in the atmosphere.

Gas-phase emissions from diesel exhaust can be oxidized in the atmosphere to form secondary PM (Gentner et al., 2017; Schauer, Fraser, Cass, & Simoneit, 2002). Diesel exhaust is an important secondary organic aerosol (SOA) precursor and source of photochemical smog. SOA from diesel exhaust and other anthropogenic sources has also been found to have a higher oxidative potential compared to such sources primary organic aerosols, and thus a potentially higher toxic potential (Pourkhesalian et al., 2015; J.; Zhou et al., 2019). While diesel fuels generally reduce the primary PM emissions, the effect on SOA yields is less known, and similar SOA yields have been found for HVO (Karjalainen et al., 2019) and FAME type fuels (Gordon et al., 2014; Jathar et al., 2017; Watne et al., 2018) relative to petroleum diesel.

Many renewable fuels have a negligible or low aromatic content compared to petroleum diesel, which has been identified as a factor that could potentially reduce the SOA formation (Gentner et al., 2017). However, as the semi-volatile organic fraction of emissions from renewable oxygen-containing fuels (FAME fuels) can be higher compared to petroleum diesel (Matti Marić, 2007; Murtonen et al., 2010; Popovicheva et al., 2017; Surawski et al., 2011), this may instead contribute to increasing the SOA formation potential per mass and possibly also to increasing the oxidative potential (Pourkhesalian et al., 2014, 2015). In addition, the role of...
lubricating oil components on SOA formation remains unclear, but has been suggested to influence the SOA formation more strongly than the fuel composition (Karjalainen et al., 2019; Le Breton et al., 2019). EATSs, composed of units such as DOCs and DPFs, significantly decrease the secondary aerosol formation of both diesel and HVO (Karjalainen et al., 2019). While the use of renewable fuels and EATSs is increasing, reliable quantitative estimates of the impact on SOA formation remains uncertain (Gentner et al., 2017).

The aim of this study was to investigate the effects of an aftertreatment system and renewable diesel fuels on primary and secondary emissions. The exhaust emissions were measured: 1) before any aftertreatment system (engine-out), 2) after a diesel oxidation catalyst (DOC) alone, and 3) after a DOC and diesel particulate filter (DPF) in series. The fuels used were Swedish MK1 ultra-low sulfur diesel and two renewable diesel fuels: HVO, and a FAME fuel (rapeseed methyl ester [RME]).

2. Method

2.1. Engine operation, fuels and exhaust aftertreatment system

The experiments were performed during steady-state operating conditions of a modern heavy-duty diesel engine modified to operate on one active cylinder (original version 6-cylinder, 13-liter engine). A version similar to the test engine is offered on the market, using the same engine hardware for diesel fuel, HVO and RME-100. The test engine was equipped with an adjustable and cooled EGR system. A low amount of EGR was used that corresponded to a combustion cylinder intake O$_2$ concentration of 18%. Throughout the experiments, the engine was operated at a constant low load (6 bar gross indicated mean effective pressure, IMEP$\text{G}$) and a speed of 1200 rpm. To simulate engine braking, the engine was additionally motored (without combustion) at the same speed (1200 rpm).

To allow a direct comparison between the emissions for the different fuels, the combustion phasing (CA50 position) was held constant at approximately 5 crank angle degrees (CAD) after the top dead center (ATDC). The effect of fine-tuning the start of injection (SOI) timing on CA50 was evaluated. Only minor differences were found and the same SOI was used for all fuels with only minor differences in injection duration. A CA50 at 5 CAD is near the optimum condition where maximum efficiency and reduced emissions can be achieved for the tested engine. An extra high-pressure injection common rail injection system supplied the fuel in a single injection at a pressure of 1200 bar through an injector with 10 holes and a 148° spray angle.

The fuels tested were Swedish MK1 ultra-low sulfur 100% petroleum diesel (B0, referred to as diesel), and two renewable diesel fuels: hydrotreated vegetable oil (HVO) and rapeseed methyl ester (RME). All fuels were tested without blending.

The EATS used was made up of two units: a diesel oxidation catalyst (DOC) and a diesel particulate filter (DPF) in a serial set-up with the DOC first. The two units were tailored for the single cylinder operation and this specific engine. The exhaust residence times inside the DOC and DPF can be considered similar to those in real vehicles. The DOC was a metallic catalyst (Pt:Pd) operating at the exhaust gas temperature, which was measured to 215 ± 6 °C for all fuels. The removal efficiency of HC and CO during the experiments confirmed that the DOC operated under adequate conditions. The DPF was a high porosity Cu–SiC model with a Pt catalyst. In modern commercial heavy-duty engines, NO$_x$ reduction techniques, for example selective catalytic reduction (SCR) using urea, is often used together with a DOC and DPF. This type of device was not included in the study; however, it is worth mentioning that SCR catalysts have been found to emit excessive ammonia (NH$_3$) which caused new particle formation of non-volatile nucleation mode particles after the DPF in some studies (Amanatidis, Ntziaziachristos, Giechaskiel, Bergmann, & Samaras, 2014; Mamakos, Schwelberger, Fierz, & Giechaskiel, 2019), but not all (Karjalainen et al., 2019).

Fig. 1. The experimental setup. Sampling was carried out before the EATS (engine-out), after the diesel oxidation catalyst (DOC) or after the DOC in combination with the diesel particulate filter (DPF). Dilution was performed with a porous tube diluter (PTD) and two ejector diluters (ED). The exhaust was either analyzed: untreated (bypass), after simulated aging with the potential aerosol mass (PAM) chamber, or after heat treatment in the thermodenuders (TD).
2.1.1. Sampling setup and dilution strategy

The aerosol sampling was performed upstream (engine-out) and downstream of the different units of the EATS, that is, after the DOC or after the DOC + DPF. The measurement setup is shown in Fig. 1. The partial flow dilution setup was composed of a porous tube diluter (PTD) with a dilution ratio (DR) of 12 and a small residence time chamber (2.5 s), in order to stabilize the nucleation mode particles. A similar setup is described by Karjalainen et al. (2019). At the applied dilution conditions, the sampling system mimics atmospheric dilution in order to simulate a particle size distribution that can be expected to form under real-world driving conditions, especially with respect to particle formation by the nucleation of vapors (Ntziaristostos et al., 2004; Rönkkö et al., 2006). After the PTD, the exhaust was further diluted with a Dekati ejector diluter with a DR of 9. The nucleation mode particles were measured after these two dilution steps with a scanning mobility particle sizer (SMPS) operating in the size range 3–60 nm (Nano-SMPS, DMA 3085 and CPC 3776, TSI Inc.), and a particle size magnifier (PSM, A11 nCNC, cut sizes 1.2–3 nm, Radiomodus), with or without a thermodenuder that has low nanoparticle losses (Heikkilä et al., 2009) operated at 265 °C. In order to monitor the dilution ratios, CO2 concentrations were measured with a non-dispersive infrared CO2 analyzer (LI-8020, LI-COR Inc.) in every step of the dilution system.

Atmospheric aging was simulated using a potential aerosol mass (PAM) oxidation flow reactor (OFR) located downstream of the 2nd dilution step. Thermodenuder (Aerodyne Inc., at 250 °C) treatment and particle measurements were performed after a 3rd dilution step with a Dekati ejector diluter (DR 4–10). The particle number size distribution was measured with a regular SMPS (Reg-SMPS) operating in the size range of 11–500 nm (DMA 3082 and CPC 3772, TSI Inc.). The equivalent black carbon (eBC) concentration was measured using an aethalometer (AE33, Magee Scientific) from the attenuation at a wavelength of 880 nm (Drinovec et al., 2015). The particle effective density and mass-mobility relationship was measured with the DMA-APM method using an aerosol particle mass analyzer (APM 3600, Kanomax) in combination with a differential mobility analyzer (DMA model 3071, TSI Inc.) and a condensation particle counter (CPC Model 3075, TSI Inc.). The organic aerosol mass concentration and composition were measured with an aerosol mass spectrometer (AMS, Aerodyne Inc.). The gas-phase emissions of NOx, HC, O2, CO, and CO2 were measured undiluted with a gas analyzer emission system (AMA 160, AVL List GmbH). Aerosol sampling for thermal-optical carbon (OC/EC) analysis was performed on undiluted exhaust on quartz filters (Pallflex Tissuequartz, 47 mm).

2.2. Aging of the emissions with an oxidation flow reactor (OFR)

The potential secondary aerosol formation upon atmospheric aging was simulated with the potential aerosol mass (PAM) OFR. The reactor consists of a 13 L steel chamber containing two Hg lamps with peak intensities at 185 and 254 nm. In this study only one of the lamps was used and operated at a reduced intensity. The lamp-mounting cylinders were continually purged with N2 to prevent O2 and OH formation, and to remove outgassing compounds. The UV light generates ozone and hydroxyl radicals (OH) that oxidize the aerosol as it moves through the chamber. The flow rate through the PAM was controlled to 5–7 L min⁻¹, resulting in an average residence time in the chamber of approximately 113–160 s. It should be noted that different parts of the aerosol experience different residence times since the flow in the reactor was neither a plug flow nor a laminar flow. The same UV light intensity was employed in all the experiments. The humidity of the aerosol prior to entering the reactor is regulated via mixing with a controlled humid air flow, and the water vapor concentration was 0.37 ± 0.02 mol m⁻³. A continuous injection of CO (40 ppm) was added to the flow to allow calculation of the OH exposure in each experiment. The concentration of CO was measured with a CO/CO2 analyzer (CO12M, Environment S.A.) while the UV lamps were on ([CO]t) and off ([CO]0). The OH exposure (OHexp) was calculated from the reaction rate constant of CO and OH (kOH+CO, Mao et al., 2009) operated at 265 °C, and the CO concentrations by Eq. (1).

\[
OH_{exp} = \frac{1}{k_{OH+CO}} \ln \left( \frac{[CO]_0}{[CO]_t} \right)
\]  

Due to variations in flow rate and OH suppression, the OH exposure varied between experiments (Table S1). The OH exposure (molecules cm⁻³ s) ranged from 3.2 × 10¹¹–1.3 × 10¹², which represents an atmospheric processing of 2.5–9.9 (4.8 ± 2.6) days if assuming an average OH concentration of 1.5 × 10⁶ molecules cm⁻³ (Mao et al., 2009). However, we observed no dependence on the yield of SOA on OH exposure in this range, which is supported by the literature that shows a low sensitivity of total SOA yields to OH exposure in this range (3–11 days) (Karjalainen et al., 2019).

2.2. Measurement techniques

2.2.1. Effective density

The particle effective density was quantified using the DMA-APM method (McMurry, Wang, Park, & Ehara, 2002). The effective density was measured at four particle mobility diameters: 58.5, 107, 196, and 296 nm. Smaller nucleation mode particles were not measured due to the lower limit of mass detection of the APM (mass detection range: 0.01 to 100 fg). Mobility size (dₐ) selection was performed with the DMA after which the APM measured the mass distribution of the selected mobility size by stepping the voltage for a constant rotating speed. The effective density, ρeff, was derived from the arithmetic mean of the measured APM voltage-number distribution and polystyrene latex (PSL) sphere reference data according to Eq. (2) (McMurry et al., 2002).

\[
ρ_{eff} = ρ_{PSL} \frac{V_{APM}}{V_{APM,PSL}}
\]

where ρPSL is the density of the PSL reference particles; VAPM is the measured arithmetic mean voltage of the sampled particles for a given mobility size and rpm; and VAPM,PSL is the theoretically calculated arithmetic mean voltage of the PSL reference particles for a
given mobility and rpm. Due to potential instrument offsets, the APM was also calibrated by measuring the $\rho_{\text{eff}}$ of PSL particles as a function of mobility diameter and comparing the measured density with the PSL manufacturer’s specifications ($1.05 \text{ g cm}^{-3}$).

2.2.2. Particle size distributions (PSDs)

The particle number size distributions were measured with two scanning mobility particle sizer systems to cover a size range of 3–500 nm. In addition, one particle size bin was measured with the PSM between 1.2 and 3 nm. Particle size distributions (PSDs) in the size range 3–60 nm were measured using the Nano-SMPS that measured close to the engine after an approx. dilution factor of 1:110. Particles size distributions in the size range 11–500 nm were determined with a regular SMPS (‘Reg-SMPS’) that measured using a dilution factor that varied between 1:500 and 1:900. Particle losses in the thermodenuders were corrected by the particle penetration curve given in Heikkilä et al. (2009), in which the penetration of 3–10 nm particles was 33–68%, and 68–75% for 11–500 nm particles. The PSDs were dilution corrected and the Nano-SMPS concentration was adjusted by comparing the measured particle concentration between 40 and 50 nm where both SMPSs have a good measurement precision. The Nano-SMPS and the Reg-SMPS PSDs were merged by attributing a weighing factor for each size bin in the overlapping size range (11–60 nm), where the Nano-SMPS precision is assumed to decrease with increasing size, and the Reg-SMPS precision to increase with increasing size. The assumed weighing factor ranged from 0 to 100% for each instrument, increasing for the Reg-SMPS (0% at 11 nm increasing up to 100% at 60 nm) and decreasing similarly for the Nano-SMPS with increasing particle size. Two modes, for the nucleation and soot modes, respectively, were fitted to the data by assuming lognormal distributions for each mode. These modes were used to extract the number and mass fractions in each mode from the size distributions.

Mass size distributions were calculated using Eq. (3), utilizing the SMPS data and DMA-APM (section 2.2.1) experimentally determined effective density ($\rho_{\text{eff}}$) as a function of electrical mobility size ($d_p$).

$$\frac{dM}{d\log d_p} = \frac{\pi d_p^3}{6} \rho_{\text{eff}}(d_p) \frac{dN}{d\log d_p}$$  \hspace{1cm} (3)

The effective density of the soot mode particles for all experiments, excluding after aging in the OFR, was assumed to follow a power law function Eq. (4) (Park, Cao, Kittelson, & McMurry, 2003), where $C$” is a constant and $D_{\text{fm}}$ the mass-mobility exponent.

$$\rho_{\text{eff}} = C”d_p^{D_{\text{fm}}-3}$$ \hspace{1cm} (4)

This was used to extrapolate a power law function for the effective density as a function of particle size of the soot mode. The nucleation mode particles were assumed to be spherical and to mainly originate from the lubricating oil (confirmed by experiments without combustion) and were hence assigned an effective density corresponding to a lubricating oil density of $\sim 0.8 \text{ g cm}^{-3}$ (Park, Kittelson, Zachariah, & McMurry, 2004). An example of the effective density and the particle number size distribution fits are shown in Fig. 2. The total mass size distribution is the superposition of the lognormal distribution fits of the mass size distributions of each mode.
up to 1000 nm (PM1). The effective density of the particles after the OFR did not follow a power law function. Instead, linear interpolations between each measurement point \(d_p: 58.5, 107, 196, \text{and} 296 \text{ nm}\) were performed for these experiments. The effective density of particles < 60 nm sampled after the OFR were assumed to be pure SOA-derived spherical particles using a constant size independent effective density that was derived from all measurements after the OFR at \(d_p = 58.5 \text{ nm} \) \(\rho_{\text{eff}} = 1.35 \pm 0.08 \frac{\text{g}}{\text{cm}^3}\). The total mass size distribution after the OFR is estimated from the lognormal distribution fit of the mass size distributions up to 1000 nm (PM1).

The soot mode mass concentrations were assessed from: 1) PM1 from the APM and SMPS (particle mass size distributions), 2) thermal-optically measured elemental carbon (EC) from filters (EUSAAR_2 protocol (Cavalli, Viana, Yttri, Genberg, & Putaud, 2010)), and 3) eBC from the aethalometer. These measurements were compared both for the primary emissions engine-out and emissions downstream of the DOC. The PM1 concentrations derived from the APM-SMPS and EC were in relatively good agreement (EC concentrations accounted for 69% ± 11% of the total soot mode mass). The eBC concentrations using the standard factory settings of the aethalometer yielded values a factor of ~2.8 ± 0.6 higher compared to the EC concentrations that were measured. This factor varied by <25% during repeated measurements (one replicate) of diesel at the engine-out sampling position. The aethalometer eBC concentration was therefore scaled to EC by dividing the original eBC concentrations from the AE33 factory settings by the empirical correction factor 2.8 derived in these experiments.

### 2.2.3. Chemical composition

The organic aerosol (OA) concentration and chemical composition of the particles were measured using a soot particle aerosol mass spectrometer (Onasch et al., 2012). The data utilized here were recorded in “tungsten only vaporization mode” (referred to as the AMS), that is, the soot module was not engaged. Data analysis was performed using PIKA 1.22A and SQUIRREL 1.62A. Calibration was performed by nebulizing and DMA-selecting 300 nm mobility equivalent diameter ammonium nitrate particles, and comparison of AMS ion rates with particle concentrations measured with a CPC according to standard procedures. OA mass was calculated by applying a standard relative ionization efficiency to nitrate (RIE\(_{\text{NO}_3}\)) of 1.4 (Alfarra et al., 2004) and using the calibrated \(\text{NO}_3\) ionization efficiency (IE\(_{\text{NO}_3}\)) of \(7.76 \cdot 10^8\) at 132 kHz.

### 2.3. Emission factors

Emission factors (g per kg burned fuel) were calculated by Eq. (5) that assumes a complete combustion (Gordon et al., 2014; Kirchstetter, Harley, Kreisberg, Stolzenburg, & Hering, 1999) and that all carbon in the fuel was combusted to CO\(_2\).

\[
\text{Emission factor (EF)} = 10^3 \frac{\Delta[\text{Pollutant}]}{\Delta[\text{CO}_2]} \frac{MW_C}{MW_{\text{CO}_2}} C_f \tag{5}
\]

where the concentration of \(\Delta[\text{Pollutant}]\) and \(\Delta[\text{CO}_2]\), are in \(\mu\text{g/m}^3\). The \(\Delta[\text{CO}_2]\) is the difference between the concentration in the exhaust pipe and in the air intake to the combustion cylinder, in order to correct for the increased CO\(_2\) in the exhaust due to the use of EGR. The carbon contents of diesel and RME were analyzed, and the carbon mass fractions \(C_f\) were 0.85 and 0.772, respectively. The carbon content for HVO was not analyzed; instead it was assumed to have the same mass fraction as diesel (Aatola, Larmi, Sarjovaara, & Mikkonen, 2008).

The emission factors per kg fuel were normalized to the energy content of the fuels and are presented in g/MJ. This was done by dividing the factors by the lower heating values of the fuels (MJ kg\(^{-1}\)). The lower heating values are 43.15 MJ kg\(^{-1}\) for Swedish ultra-low sulfur diesel, 44.1 MJ kg\(^{-1}\) for HVO (Engman et al., 2015), and 37.3 MJ kg\(^{-1}\) for RME (Thuijl, Roos, & Beurskens, 2003).

Fig. 3. The emission factors of NO\(_x\) (left axis), HC and CO (right axis) for all fuels measured engine-out (a). In (b) the emission factors were compared for engine-out and downstream the DOC for diesel only. The error bars represent ±1 std. dev. of repeated measurements \((n = 2-6)\).
3. Results and discussion

3.1. Primary NO\textsubscript{x}, HC, and CO emissions

The effects on NO\textsubscript{x}, HC and CO emission factors from substituting petroleum diesel with HVO and RME are presented in Fig. 3a. Substitution with RME had a larger impact on the CO and HC emissions than substitution with HVO. While RME usage reduced the HC and CO emissions by 67% (95% CI [63%, 72%]) and 31% (95% CI [26%, 35%]), respectively, the HVO usage had a lower impact relative to diesel. Substitution with HVO reduced the HC emissions by 20% (95% CI [15%, 25%]) and had a smaller impact on the CO emissions, which were reduced by 9% (95% CI [7%, 12%]) compared to diesel.

Compared to diesel, the small increase in NO\textsubscript{x} emissions observed in this study was not statistically significant for neither HVO nor RME (unpaired t-test, \( p = 0.44 \) for HVO, \( p = 0.064 \) for RME). In this study, the engine was operated with the same ignition delay for all fuels and similar short premixing times. This means that the observed difference in emissions is not due to variations in premixing time due to the different cetane numbers of the tested fuels, which is one important mechanism that can change the emissions for diesel fuels (Giakoumis et al., 2012; Tree & Svensson, 2007).

The decrease in HC and CO emissions were expected and followed previously reported observations of reduced emissions of FAME type biodiesels (Giakoumis, Rakopoulos, Dimaratos, & Rakopoulos, 2012; Horn, Egnell, Johansson, & Andersson, 2007; Georgios Karavalakis et al., 2017) and HVO (George Karavalakis et al., 2016; Kuronen et al., 2007). Most previous studies of FAME type fuels report a slight increase in NO\textsubscript{x} emissions (reviewed in Giakoumis et al., 2012; Magín; Lapuerta, Armas, & Rodríguez-Fernández, 2008), while for HVO, both no change (Shukla et al., 2018) and an increase or decrease depending on engine load (George Karavalakis et al., 2016; McCaffery, Karavalakis, Durbin, Jung, & Johnson, 2020; Zubel et al., 2016) have been reported.

The oxygen content of RME can cause a higher oxygen entrainment in the premixing, which would increase the completeness of the combustion and thus decrease the HC and CO emissions (as well as the PM emissions) with a trade-off of slightly higher temperatures

![Fig. 4.](image-url)
and increased NOx emissions (M. Lapuerta, Armas, & Rodríguez-Fernández, 2008; Mueller, Boehman, & Martin, 2009). For HVO, the similarities to petroleum diesel in physicochemical properties of the fuel—such as viscosity, lower heating value, heat of vaporization, and stoichiometric air to fuel ratio—are hypothesized to cause similar in-cylinder temperatures as diesel and hence similar NOx emissions (Shukla et al., 2018). Differences such as higher H/C ratio and the absence of aromatics in HVO compared to diesel are hypothesized to cause the lower HC and CO emissions (McCaffery et al., 2020; Pflaum, Hofmann, Geringer, & Weissel, 2010). The reductions of HC and CO were somewhat lower than previously reported (Bohl, Smallbone, Tian, & Roskilly, 2018; McCaffery et al., 2020; Napolitano et al., 2015; Pflaum et al., 2010), when HVO has been compared to standard diesel with >15% aromatic content. The aromatic content of the Swedish MK1 diesel is <5% which could be the reason for the relatively lower reductions.

The effects of employing the DOC for the case with petroleum diesel are presented in Fig. 3b. The DOC effectively reduced the engine-out HC and CO emission factors by 73% (95% CI [67%, 79%]) and 80% (95% CI [78%, 81%]), respectively. The reduction of HC by the DOC was similar to the reduction caused by a fuel change to RME during engine-out. The DOC had a low influence on the NOx emissions (5% reduction, 95% CI [0%, 11%]). The DOC converts NO to NO2, but the individual concentrations of NO and NO2 were not measured. The conversion to NO2 can be important as DPFs utilize NO2 for continuous regeneration (soot removal).

### 3.2. Primary particle emissions

#### 3.2.1. Particle size distributions

Fig. 4a–d shows the number and mass weighted particle size distributions (3–500 nm) of emissions engine-out and downstream the DOC, as well as for motoring (no combustion). Nanoparticle emissions in the interval 1.2–3 nm particle size (measured with the PSM) are also indicated with horizontal bars. The particle number size distributions were characterized by a nucleation mode and a soot mode, both in the engine-out exhaust (Fig. 4a) and downstream the DOC (Fig. 4b). The emissions during motoring (no combustion) were characterized by a single nucleation mode and no soot mode. Lognormal distributions were fitted to the two modes (SMPS only, example in Fig. S2), and the geometrical mean diameter (GMD), geometrical standard deviation (GSD), and number concentrations for each mode are summarized in Table 1. The DPF reduced particle number emissions by 4 orders of magnitude for all fuels. Because of the high dilution ratio used (~10^3), size distributions after the DPF could not be resolved.

Nucleation mode particles were strongly reduced by the DOC for all tested fuels (Table 1). The DOC reduced the nucleation mode particle number concentrations by about one order of magnitude, and the mobility diameters shifted towards smaller sizes compared to the engine-out position (Table 1, Fig. 4a and b). This was also seen in the relatively high concentrations observed by the particle size magnifier (PSM) downstream the DOC compared to the engine-out situation. Similar nucleation mode reductions of diesel exhaust from a light-duty diesel engine was found by Zeraati-Rezaei et al. (2020) who attributed this to decreased concentrations of intermediate and semi-volatile organic compounds. However, their engine was operated with higher EGR which led to a higher soot mode from a light-duty diesel engine was found by Zeraati-Rezaei et al. (2020) who attributed this to decreased concentrations of intermediate and semi-volatile organic compounds. 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The emissions during motoring (no combustion) were characterized by a single nucleation mode and no soot mode. Lognormal distributions were fitted to the two modes (SMPS only, example in Fig. S2), and the geometrical mean diameter (GMD), geometrical standard deviation (GSD), and number concentrations for each mode are summarized in Table 1. The DPF reduced particle number emissions by 4 orders of magnitude for all fuels. Because of the high dilution ratio used (~10^3), size distributions after the DPF could not be resolved.

**Table 1**

| Fuel | Engine-out | DOC | DOC + DPF |
|------|------------|-----|----------|
| HVO  | 1.05 x 10^14 | 7.16 x 10^13 | 1.13 x 10^10 |
| RME  | 1.17 x 10^14 | 1.04 x 10^13 | 9.34 x 10^10 |

The particle number concentration and size for all fuels measured engine-out, downstream the DOC, and DOC + DPF. The geometric mean diameter (GMD), geometric standard deviation (GSD), and number emissions (per MJ) of the nucleation mode (N*) and soot mode (S*) were estimated from lognormal fits to the bimodal particle number size distributions (example in Fig. S2). The number concentration of the 1.2–3 nm particles was measured by the PSM. The total particle number concentration was calculated from the original SMPS data (3–500 nm). No modes could be fit from the particle size distributions measured downstream the DOC + DPF. The conversion factors for the emissions factors are given.
Previous studies on various fuels and diesel engines have indicated that a significant fraction of the nucleation mode of exhaust particles originates from the lubricating oil (de Filippo & Maricq, 2008; Matti Maricq, 2007; Tobias et al., 2001). The lubricants from the moving engine parts have been hypothesized to escape into the combustion chamber with the piston movements and exit with the exhaust emissions. Oil can thus be directly atomized or alternatively, nucleate or condense from oil vapors during cooling and dilution of the exhaust emissions. Together with the similarity in nucleation mode geometric mean diameter (GMD) during motoring and combustion with diesel, HVO, and RME, the results show that the main fraction of the nucleation mode is lubricant derived.

The PSM extends particle size analysis below the detection limit of the Nano-SMPS. The PSM results indicated in Fig. 4a and b suggest that the nucleation mode is broader than could be resolved by the Nano-SMPS. The observation of nonvolatile nanoparticles (down to 1.2 nm) downstream the DOC or TD (Fig. 4b) provide evidence that a significant fraction of the nucleation mode particles had solid or nonvolatile cores onto which more volatile organic material had condensed.

The relationship between particle mobility size and mass was determined with the DMA-APM system and was used in conjunction with the number size distributions to calculate the particle mass size distributions presented in Fig. 4c and d. The effective density decreased with increasing size for all studied particle sizes \( d_p = 58.5, 107, 196, 296 \) nm (Fig. S1). The mass-mobility exponent, \( D_{fm} \) (Eq. (4)), varied between 2.26 and 2.42 for the fuels at engine-out and between 2.33 and 2.41 downstream the DOC. RME had the highest values of \( D_{fm} \) and HVO had the lowest values for both sampling positions. Because the nucleation mode was assumed to be dominated by lubricating oil, the material density of the lubricating oil \((0.8 \text{ g cm}^{-3})\) was used as the effective density for nucleation mode particles. HVO had the largest nucleation mode mass. The soot mode mass was lower relative to petroleum diesel for both HVO and RME (Fig. 4c). During motoring, no clear soot mode was present and >90% of the mass was in the nucleation mode. The DOC effectively removed the nucleation mode mass (Fig. 4d, Table S2) compared to engine-out for all fuels and during motoring, confirming that the nucleation mode mass was dominated by volatile lubricant-derived particles. The mass GMD of the soot mode was unaffected by the DOC for diesel and HVO but increased from 136.5 nm to 157.1 nm for RME (Table S2).

For all fuels, the measured effective densities of the soot mode downstream the DOC were lower compared to the engine-out (Fig. S1). This indicates that the soot agglomerates became more porous due to the loss of volatile material in the DOC that would otherwise condense onto the soot agglomerates during cooling and dilution. Both the effective density and mass-mobility exponent of particles in the soot mode were slightly higher for RME compared to diesel and HVO. We hypothesize that this was due to differences in the particle morphology. FAME fuels may generate more compact soot aggregates because of stronger soot oxidation that preferentially removes the longest branches from the aggregates. Another explanation could be that the soot agglomerates from RME contain a higher fraction of organic compounds. However, this was contradicted by the measurements of organic aerosol (OA measured with AMS) and eBC presented in the next section.

### 3.2.2. Chemical composition and mass emission factors

As shown by the size distributions, the nonvolatile particle mass emissions were dominated by a soot mode that primarily consists of elemental carbon, which is closely related to the climate relevant black carbon (BC) emissions. Equivalent BC (eBC) was measured continuously with an aethalometer, and eBC emission factors in comparison to diesel were reduced by 48% (95% CI [28%, 69%]) for HVO and 50% (95% CI [26%, 74%]) for RME (Fig. 5). The decrease of eBC emissions observed here is related to the reduced soot mode number concentration (Fig. 4c, Table 1) and is similar to our previous study with higher EGR levels using the same fuels in a similar engine set-up (Gren et al., 2020). Similar eBC reduction potentials have also been reported for HVO by others (Pflaum et al., 2010; Zubel et al., 2016). The driving mechanism for the reduction could be a combination of many parameters. The main hypotheses are:

- **Hypothesis 1:** The decrease in eBC emissions is due to a lower soot agglomeration efficiency, resulting in smaller soot aggregates. This could be caused by differences in the particle morphology, with FAME fuels generating more compact soot aggregates due to stronger soot oxidation that preferentially removes the longest branches from the aggregates.
- **Hypothesis 2:** The lower soot agglomeration efficiency may be due to a higher fraction of volatile organic compounds in the soot agglomerates from RME, which preferentially condense onto the soot aggregates during cooling and dilution.
- **Hypothesis 3:** Other parameters, such as differences in the processing of the fuels or the engine operation, may also contribute to the observed reduction in eBC emissions.

![Fig. 5. Equivalent black carbon (eBC) emission factors in mg MJ$^{-1}$ for all tested fuels engine-out, downstream the DOC and downstream the DOC + DPF. Error bars represent ±1 std. dev. The eBC values were divided by the empirically determined correction factor 2.8 to be scaled to EC (Sect. 2.2.2).](image-url)
the absence of aromatic content which act as soot precursors (McCaffery et al., 2020; Singh, Subramanian, & Singal, 2015; Tan et al., 2013; Wu, Huang, Zhang, Fang, & Yin, 2007), and ii) the more homogenous composition and resulting lower boiling temperature of HVO compared to diesel and RME, which cause improved spray characteristics and mixture formation which reduce the soot formation (Bhardwaj et al., 2014). The eBC reduction for RME is instead likely related to the oxygen content in the RME fuel which has been hypothesized to both increase the completeness of the combustion in local regions of the combustion cylinder and increase the soot oxidation during the late cycle combustion (Magín Lapuerta, Armas, & Rodríguez-Fernández, 2008; Matti Maricq, 2007). The DOC alone had no significant impact on the eBC emission factors, while eBC emissions were almost absent with the addition of the DPF (after the DOC and DPF in combination). We calculated the average eBC removal efficiency after the DPF to 99.4 ± 0.5%. The variability of eBC emission factors was <10% during replicate measurements on diesel (engine-out) and HVO (DOC).  

Fig. 6 shows the mass emission factors of equivalent black carbon (eBC), organic aerosol (OA) and the total particle mass calculated from the particle size distributions (PM1). The nucleation mode mass fractions of PM1 during engine-out were 18 ± 2% for diesel, 30 ± 3% for HVO, and 18 ± 2% for RME. The DOC reduced the nucleation mode mass fractions to <1% for all fuels. The effect of the DOC on the soot mode mass emission factors was small, with an average reduction of 12 ± 13%. The average eBC concentrations were unaffacted by the DOC. We hypothesize that the small mass reduction after the DOC was due primarily to the removal of organic aerosol that was likely (due to the unimodal shape of APM mass distributions) internally mixed with the soot mode eBC rich particles. The effective density of the soot mode was found to decrease after the DOC (Fig. S1). This supports that internally mixed organic components were removed in the DOC, which was confirmed by the OA mass emissions which decreased for all fuels by on average 45 ± 11% after the DOC (Fig. 6). The OA was measured by the AMS, which has a very low collection efficiency of \( d_{p} < 50 \text{ nm} \); hence, the OA that was measured should primarily originate from the soot mode. The soot mode mass, calculated from the PSDs and the effective densities, correlates reasonably well (by a factor 1.3 higher) with the sum of the measured eBC (scaled to EC by a factor 2.8, Sect. 2.2.3) and OA mass. This also suggests that the OA fraction measured by the AMS captures the organic fraction of the soot mode, but not the organic mass in the nucleation mode.

Looking at the effects of fuel on nucleation mode and OA emissions, we found that HVO emitted the highest nucleation mode mass followed by diesel and RME. For OA measured with the AMS, representing the OA of the soot mode, diesel had the highest emissions followed by HVO and RME. We hypothesize that the difference between diesel and HVO is explained by the higher condensation sink of the soot mode for diesel (Fig. S4), which leads to a higher fraction of the condensable organics ending up in the soot mode. We note that the gas-phase HC emissions were similar for diesel and HVO but three times lower for RME. The results in Fig. 6 show that RME was associated with both lower nucleation mode mass and OA emissions in the soot mode. This suggests that particle phase organic species were significantly reduced for RME compared to the other two fuels in similar fashion as gas-phase HC emissions. The DOC drastically reduced the nucleation mode mass emissions, but the effect on OA was more modest (average 45% reduction across fuels). We hypothesize that this is because the DOC removes the majority of the condensable mass (Bhardwaj et al., 2014). This in turn leads to a reduction of supersaturations during sampling, thereby suppressing nucleation and early growth by condensation. This results in a much smaller condensation sink provided by the nucleation mode, thus shifting the condensation after the DOC towards the soot mode on a relative basis, explaining the lower effect of the DOC on OA measured by the AMS compared to nucleation mode mass.

Fig. 6. The mass emission factors of equivalent black carbon (eBC), organic aerosol (OA measured with the AMS), and soot and nucleation modes. The masses of the nucleation and soot mode were derived from particle mass size distributions (Sect. 2.2.2). The error bars show ±1 relative std. dev. of the summed mass of either the eBC and OA, or soot and nucleation modes (PM1), based on replicates of diesel and HVO after the DOC. The eBC values were divided by the empirically determined correction factor 2.8 to be scaled to EC (Sect. 2.2.2).
3.3. Secondary particle emissions

The particle number size distributions and effective densities of the engine-out emissions after simulated atmospheric aging in the OFR are shown in Fig. 7a and b. The shape of the particle size distributions changed upon aging, and a new mode appeared in between the former nucleation and the soot mode (Fig. 7a). After aging, the effective density of the 58.5 nm particles increased to 1.32–1.37 g cm\(^{-3}\) regardless of fuel (Fig. 6). This is similar to the literature values of SOA density (Kostenidou et al., 2007; Nakao et al., 2013), indicating that these particles were primarily comprised of SOA. The largest increase in effective density upon aging was for the 107 nm particles which increased from 0.70 to 1.17 g cm\(^{-3}\) (HVO), and from 0.73 to 1.13 g cm\(^{-3}\) (diesel), respectively (see Fig. 7b). A similar large increase in effective density upon aging at this size was not observed for RME, for which the initial soot mode still dominated at this size. Aged RME emissions instead showed a higher increase in effective density compared to the other fuels at 196 nm. At \(d_p > 200\) nm for diesel and HVO (no data on RME), the effective density of the soot mode only increased slightly after aging from the initial values of around 0.4, which is far below the SOA density of \(\sim 1.35\) g cm\(^{-3}\). The small increase in effective density was attributed to the condensation of only minor fractions of SOA onto the soot mode particles.

The particle mass size distributions engine-out and after the DOC after aging are shown in Fig. 8a and b. The aged PM emissions after the DOC + DPF could not be clearly resolved from the SOA background (Fig. S5c). After aging, the new mode dominated the mass emissions for all fuels engine-out (Fig. 8a) and the bimodality of the particle size distribution was not as clear as before aging. This latter observation is likely due to condensational growth primarily on existing nucleation mode particles. By combining with the DMA-APM observations, we hypothesize that the aged emissions engine-out consist of two particle types: 1) SOA dominated particles that grow due to SOA condensation on existing nucleation mode particle (these particles dominate the mass size distribution) and 2) the soot mode present from the engine-out conditions with only minor fractions of condensed SOA. Notably, the secondary aerosol mass formation was lower compared to diesel for both renewable fuels, with the largest reduction seen for RME as is clearly illustrated in Fig. 8a.

Aging of the exhaust after the DOC had a much lower effect on the mass size distribution (Fig. 8b) compared to aging of engine-out emissions, and only a weak increase similar to the new mode can be seen. Perhaps a larger fraction of the low amounts of formed SOA ended up on the soot mode particles, as the relative surface area of the nucleation mode compared to soot mode particles is lower in this case. We note that these size distributions may not accurately describe real-world aged emissions size distributions since the aging in the reactor speeds up the process and the aerosol dynamics (i.e., physics) may be different compared to the atmosphere.

The mass emission factors of fresh and aged exhaust, estimated from the sum of eBC and OA or mass weighted size distributions (PM1) are shown in Fig. 9a and b. The total aged engine-out PM1 mass emissions were in comparison to the primary emission factors about 5 times higher for diesel, 4 times higher for HVO, and 3 times higher for RME (based on PM1 from the SMPS and effective density derived mass distributions). A strong reduction of aged mass formation of PM1 was observed after the DOC for all fuels. Significant reductions in SOA formation of diesel exhaust after similar aftertreatment systems have been reported in previous studies (Chirico et al., 2010; Karjalainen et al., 2019). Compared to primary emissions, the aged PM1 mass was less than a factor 2 times higher for all fuels downstream the DOC, while the summed aged mass emissions of eBC and OA were even less affected. The eBC + OA consistently showed much lower mass concentrations than the PM1 for the aged emissions engine-out. This may similarly to that described above for the fresh emissions, be due to the reduced collection efficiency of the AMS at particle sizes below 100 nm where most of the aged PM exists.

The total gas-phase HC emissions have been identified as a key parameter in secondary aerosol formation (Karjalainen, Timonen, et al., 2016; Roth et al., 2020). The HC emission factors were significantly lower compared to diesel for RME (67% reduction) and...
slightly reduced for HVO (20% reduction) (Fig. 3a). The DOC was found to reduce HC emission by 73% for the diesel combustion (Fig. 3b). In Fig. S6a we show that the HC emissions correlated with aged PM1 mass emissions ($r = 0.96$), supporting that the reduced HC emissions likely explain the lower aged PM emissions. In addition, it has been suggested that the SOA formation of diesel emissions depends on the aromatic content in the fuel (Gentner et al., 2017), and HVO and RME are both aromatic-free fuels. The reduced SOA emissions of RME and HVO compared to diesel may therefore potentially be understood as a combination of reduced HC emissions and aromatic-free fuel. For gasoline engines oxygenated fuels (higher blend of ethanol) have been found to reduce the SOA formation. This was attributed to decreased HC emissions (Roth et al., 2020) and an increased fraction of short-chained organic species (Timonen et al., 2017). In this study the oxygen content of RME caused the decreased HC emissions which we hypothesize is also the main cause of the decreased SOA formation.

The lubricating oil has also been found to influence the secondary aerosol formation independent of fuel in diesel engines (Karjalainen et al., 2019; Le Breton et al., 2019; Watne et al., 2018) The DOC was efficient at removing the nucleation mode particles and hence the lubricant-derived and organic components of the engine-out emissions. This could have further contributed to the reduced secondary PM emission factors (Fig. S6b). However, the effect was likely small in comparison to the fuel-induced effect of altered HC emissions that is illustrated in Fig. S6a.

The emission factors of the total aged PM after the DOC are upper limit values in this study (Fig. 9). By measuring the SOA formation potential of the dilution air, the OFR background contribution could be as high as 60% of the total aged PM mass after the DOC (mass estimated from the mass size distribution of background measurements), while relatively much smaller compared to aged engine-out emissions (20%), see (Fig. S5). Hence, the SOA mass emissions reported after the DOC may be biased high compared to engine-out.
3.4. Limitations

We carried out a detailed study on the fundamental aerosol properties of the exhaust emissions from various exhaust aftertreatment combinations (DOC, DOC + DPF) and fuel choices (diesel, HVO, RME) for a heavy-duty diesel engine operated under low-load (IMEP\(_g\) = 6 bar) at steady conditions. The engine was also operated in motored mode to simulate engine breaking. The resulting particle emissions and properties cannot be generalized to all applications of this engine type in real-world use. This study does not cover the effects on aerosol emissions from, for example, variance in engine load and transients in operation, different effective temperatures of the DOC and DPF, specific emissions after engine cold start, or specific DPF regeneration events.

The secondary aerosol formation was simulated with the PAM OFR, which has been shown to yield SOA formation similar to smog chambers when operated under similar conditions. However, both can underestimate the yields compared to the SOA found in the atmosphere (Bruns et al., 2015). Hence, it is not straightforward to estimate the atmospherically relevant absolute SOA mass emissions directly from these observations, since the emissions may be biased low for a number of reasons. SOA yields may be biased low due to wall-losses and SOA precursor fragmentation caused by the intense oxidation in the OFR (Lambe et al., 2015). However, aerosol with high particle surface area entering the OFR, such as in our experiments (2.4–4.0 × 10^4 μm^2 cm⁻³, see Fig. S4), have been found to cause conditions where the wall-losses and fragmentation should be small compared to the mass condensed onto existing particles (Ahlberg, Eriksson, Brune, Roldin, & Svenningsson, 2018). Our measured eBC concentrations were similar before and after aging (Fig. 9), which indicates low particle wall-losses, at least for the soot mode particles. In a similar OFR chamber, particle losses increased from 25% at 50 nm to up to 60% for 10 nm particles at a flow rate of 10 L min⁻¹ (Karjalainen, Timonen, et al., 2016). In this study we used a lower flow rate (5–7 L min⁻¹) that may have further increased the losses of small particles. Consequently, absolute number concentrations after aging are not considered. In addition, parameters such as temperature and relative humidity may significantly influence the SOA yields (e.g. Tillmann et al., 2010; Zhou et al., 2011), and in our oxidation flow reactor, the temperature was relatively high (>28 °C) and the RH relatively low (20–25%), which may have biased the SOA yields further low relative to more typical atmospheric conditions (Tillmann et al., 2010; Zhou et al., 2011). While the absolute emissions are somewhat uncertain, we expect the relative differences in SOA emission factors between the respective fuels to remain robust for a range of atmospheric conditions, since the respective biases are expected to be similar for the different experiments.

4. Conclusions

Renewable diesel fuels and exhaust aftertreatment are used to reduce greenhouse gas emissions and improve air quality by reducing primary emissions. However, for the total atmospheric impact the secondary emissions are also of great importance, which from renewable diesel fuels and from different emission reduction techniques are less known. In line with previous studies, many of which have been reviewed (Knothe, 2010; Knothe et al., 2015; Mahmudul et al., 2017), we found that replacing petroleum diesel with renewable diesel fuels reduced the primary particle emissions and important gas-phase emissions. We found that the engine-out eBC emissions, when normalized to the fuel energy content, were reduced by 48% for HVO and 50% for RME in comparison to petroleum diesel. The reductions in eBC induced by the fuel substitutions were similar after the introduction of the DOC, which was the first in line of the exhaust aftertreatment units included in the study. Engine-out CO and HC emissions were strongly reduced for RME in comparison to both diesel and HVO. The RME reductions in comparison to diesel were 67% (95% CI [63%, 73%]) for HC, and 31% (95% CI [26%, 35%]) for CO. The DOC efficiently reduced the emissions of CO (by 80%), of gas-phase HC (by 73%), and of nucleation mode particles by both number and mass. The DPF additionally reduced eBC and the total PM mass emissions by >99%.

The total mass emissions were estimated using two independent methods: 1) from integrating the SMPS mass weighted size distributions using empirically determined effective densities, and 2) from the sum of eBC and OA (measured by the AMS). The two mass emission estimates were qualitatively in good agreement for the soot mode. The lubricant-originated nucleation mode contributed up to 30% of the total mass, and nonvolatile particles existed down to at least 1.2 nm in particle size. The AMS did not capture the organics in the nucleation mode due to its low collection efficiency of particles below 50 nm (Jayne et al., 2000). The decrease in the particle effective densities after the DOC shows that the organic coating material of soot mode particles was removed in the DOC (increased porosity of the soot agglomerates) for all fuels. The increase in effective densities after aging in the OFR (increasing asymptotically towards the SOA density) indicates that the majority of the SOA condensed on existing nucleation mode particles and to a lesser extent on pre-existing soot mode particles.

Without aftertreatment, the secondary aerosol mass emission factors (per MJ) were 3–5 times higher than the primary emission factors at the engine setting that was investigated. With the DOC, the increase in mass emissions did not exceed a factor 2. It was hypothesized that the secondary PM yield upon aging in the OFR depended both on the aromatic content of the fuel and the concentration of emitted gas-phase HC. RME generated lower HC emissions and showed a much lower potential secondary PM formation compared to both diesel and HVO. Our results indicate that RME may not only reduce the primary emissions, but also significantly reduce the secondary aerosol mass formation. This is contrary to some previous studies on FAME fuels (Jathar et al., 2017; Watne et al., 2018) that showed slightly increased SOA yields, which calls for further research to improve our understanding of the components that drive the secondary aerosol formation from renewable fuels. The identification of renewable and sustainable fuels that reduce primary emissions and delayed secondary emissions is a particularly important objective to mitigate CO₂ emissions and simultaneously improve air quality with the existing vehicle fleet. This study shows that renewable fuels similar to those examined here can have important potential for reducing both primary and secondary PM emissions in already existing vehicles, especially from those not equipped with aftertreatment systems.
Author contribution

Conceptualization: PK, VBM, JP, MT, LG. Formal data analysis, interpretation and visualization: LG. Engine operation and data analysis: MN, SS. Design of dilution systems, Nano-SMPS and PSM operation and data analysis: LM, PK. Collection of AMS data and interpretation: VBM. AMS data analysis and interpretation: ACE. APM operation and data analysis: JF, TBK. Design of OFR experiments and data analysis: PK, BS. LG prepared the original manuscript, with contributions from VBM and JP. All co-authors have critically reviewed and approved the manuscript.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

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