Absorption and photoconductivity spectra of amorphous multilayer structures

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Abstract
The experimental results regarding optical absorption and steady-state photoconductivity of amorphous single-layer structures (Al–As_{0.40}S_{0.30}Se_{0.30}–Al, Al–Ge_{0.09}As_{0.09}Se_{0.82}–Al, and Al–Ge_{0.30}As_{0.04}S_{0.66}–Al) and of an amorphous heterostructure (Al–As_{0.40}S_{0.30}Se_{0.30}/Ge_{0.09}As_{0.09}Se_{0.82}/Ge_{0.30}As_{0.04}S_{0.66}–Al) at different values of the voltage, with positive or negative polarity, applied to the illuminated top Al electrode are presented and discussed. The complex structure of the photocurrent spectra is attributed to the different values of the optical bandgap of the involved amorphous layers (\(E_g \approx 2.0\) eV for As_{0.40}S_{0.30}Se_{0.30} and Ge_{0.09}As_{0.09}Se_{0.82} and \(E_g \approx 3.0\) eV for Ge_{0.30}As_{0.04}S_{0.66}). The obtained experimental results are discussed taking into account the light absorption depending on the nature and the thickness of each amorphous layer, on the wavelength, and on contact phenomena at the interfaces between different layers and between the amorphous layers and the metal electrodes with different work functions.

Introduction
The As–S–Se, Ge–As–Se, and Ge–As–S ternary glass systems currently attract a lot of attention because of their wide application in IR optics, non-linear optics, photonics, optoelectronics, and as recording media for holography and e-beam lithography [1-3]. The physical properties of covalently bonded glasses are determined by the mean coordination number \(Z\) (average number of covalent bonds per atom) [4]. It is well known that the functionality of many photonic and optoelectronic devices is based on the intrinsic photovoltaic effect. The photocurrent spectra and the kinetics of the photocurrent can provide information regarding the mechanisms of generation, recombination, and drift processes of non-equilibrium carriers in amorphous materials. Thus, investigations of stationary and transient characteristics of the photoconductivity of ternary amorphous thin films are of special interest. For thermally deposited amorphous films, the structure of which exhibits a higher level of disorder than that of the bulk glasses, the incorporation of impurity atoms is easier, and in many cases the metal additives could...
become electrically active. The influence of Sn impurities on stationary and transient photocconductivity was demonstrated for amorphous As$_2$Se$_3$Sn$_x$ thin films [5]. The introduction of Sn in the host material increases the drift mobility and the photosensitivity of the amorphous material. According to $^{119}$Sn Mössbauer spectroscopy studies of the As$_2$Se$_3$:Sn$_x$ glassy system [6] and X-ray photoelectron spectroscopy studies of As$_x$Ge$_y$Se$_{1−2x}$ glasses [4], the introduction of elements such as Sn or Ge in glasses based on arsenic selenides, leads to the formation of new tetrahedral Sn(Se$_{1/2}$)$_4$ and quasi-octahedral SnSe structural units or of GeSe$_4$, respectively. Recently, experimental results of steady-state and transient photocurrents of amorphous Ge$_x$As$_{1−x}$Se$_2$ and (As$_2$Se$_3$)$_{1−x}$Sn$_x$. thin films were presented [7,8]. It was established that the predominant mechanisms of recombination of the photo-excited carriers in the investigated amorphous materials are mono- and bimolecular and that the transport is controlled through multiple trapping processes with exponential distribution of the localized states in the bandgap. Regarding the physics and applications of chalcogenide materials, multilayered amorphous thin-film structures are especially interesting because they offer advantages over single-layer structures [9,10].

**Experimental**

The bulk chalcogenide glasses As$_{0.40}$S$_{0.30}$Se$_{0.30}$, Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$, and Ge$_{0.30}$As$_{0.04}$Se$_{0.66}$ were prepared from the elements (Ge, As, S, and Se; 99.9999% purity) by conventional melt quenching. The initial materials were weighed and mixed with the elements (Ge, As, S, and Se; 99.9999% purity) by conventional melt quenching. The initial materials were weighed and mixed with the elements (Ge, As, S, and Se; 99.9999% purity) by conventional melt quenching. The initial materials were weighed and mixed with the elements (Ge, As, S, and Se; 99.9999% purity) by conventional melt quenching. The initial materials were weighed and mixed with the elements (Ge, As, S, and Se; 99.9999% purity) by conventional melt quenching. The initial materials were weighed and mixed with the elements (Ge, As, S, and Se; 99.9999% purity) by conventional melt quenching. The initial materials were weighed and mixed with the elements (Ge, As, S, and Se; 99.9999% purity) by conventional melt quenching. The initial materials were weighed and mixed with the elements (Ge, As, S, and Se; 99.9999% purity) by conventional melt quenching. 

Transmission spectra  $T = f(\lambda)$ of the separate amorphous thin films Ge$_{0.30}$As$_{0.04}$Se$_{0.66}$ (1), Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$ (2), As$_{0.40}$S$_{0.30}$Se$_{0.30}$ (3), and the HS As$_{0.40}$S$_{0.30}$Se$_{0.30}$/Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$/Ge$_{0.30}$As$_{0.04}$Se$_{0.66}$ (4). The thin film layer Ge$_{0.30}$As$_{0.04}$Se$_{0.66}$ with the largest bandgap energy, $E_g \approx 3.0$ eV [11], which was placed on the top of the multilayer structure, has a thickness of $d \approx 200$ nm and was transparent to the incident visible light to reach the other layers with a bandgap energy of $E_g \approx 2.0$ eV [12,13] and with a thicknesses of $d \approx 500$ nm for Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$ and $d \approx 1000$ nm for As$_{0.40}$S$_{0.30}$Se$_{0.30}$. Figure 2 shows that the amorphous film Ge$_{0.30}$As$_{0.04}$Se$_{0.66}$ is highly transparent to incident light in the visible region, in contrast to the other thin-film structures (curve 4).

**Results and Discussion**

Figure 1 shows the transmission spectra  $T = f(\lambda)$ of the separate amorphous thin films and multilayer HS (1: Ge$_{0.30}$As$_{0.04}$Se$_{0.66}$, 2: Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$, 3: As$_{0.40}$S$_{0.30}$Se$_{0.30}$, 4: As$_{0.40}$S$_{0.30}$Se$_{0.30}$/Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$/Ge$_{0.30}$As$_{0.04}$Se$_{0.66}$). Figure 2 shows the function $\alpha = f(hv)$ of the absorption coefficient for single amorphous layers and for the amorphous HS. The absorption coefficient $\alpha$ for each layer can be calculated using the expression:

$$\alpha = \frac{1}{d} \ln \left( \frac{1-R}{T} \right),$$

(1)

$\alpha$ is the absorption coefficient, $d$ is the thickness of the film, $R$ is the reflectance, and $T$ is the transmittance.
The absorption coefficient $\alpha$ of each separate component layer is higher than that of the HS. The highest value of the absorption coefficient $\alpha$ was obtained for the amorphous film of $\text{Ge}_{0.30}\text{As}_{0.04}\text{S}_{0.66}$ with the smallest thickness of $d \approx 200$ nm. The amorphous structures described above with two Al electrodes, of which the upper one was transparent to the incident light, have been used for electrical and photoelectrical measurements.

Figure 3 and Figure 4 show the current–voltage ($I$–$V$) characteristics of the amorphous thin-film HS $\text{Al–As}_{0.40}\text{S}_{0.30}\text{Se}_{0.30}/\text{Ge}_{0.09}\text{As}_{0.09}\text{Se}_{0.82}/\text{Ge}_{0.30}\text{As}_{0.04}\text{S}_{0.66}$ with positive or negative voltage applied to the top Al electrode in the dark (curve 1) and under illumination (curve 2).

When a positive voltage is applied to the top Al electrode, a sharp increase of the current occurs, only under illumination, when the applied external electric field reaches high values. When a negative voltage is applied, this sharp increase of the current occurs in dark and under illumination. This effect was observed for amorphous thin-film structures with different electrodes ($\text{As}_2\text{S}_3$, $\text{As}_2\text{S}_3\text{Ge}_x$) and amorphous HS ($\text{As}_2\text{S}_3/\text{Sb}_2\text{S}_3$, $\text{Si/As}_2\text{S}_3$) [14]. $I$–$V$ characteristics are described by the expression $I = AV^m$, where $A$ is a constant coefficient, and $m$ is the exponent. When carriers are injected from an amorphous or crystalline layer (or electrode) into another amorphous layer, $m$ cannot take values larger than 1. The part of the $I$–$V$ characteristics with $m = 2$ is explained by the existence of an exponential distribution of the localized sates in the bandgap of the amorphous material. In this case the $I$–$V$ characteristics are described by the expression [15]:

$$j = N_e\mu e\epsilon N_e^d \left( \frac{V^m}{d^{d-1}} \right),$$

(2)

where $j$ is the current density, $N_e$ is the effective density of states in the valence band, $N_i$ is the concentration of traps, $e$ is the electron charge, $\mu$ is the drift mobility, $\epsilon$ is the dielectric constant, $\tau = \Delta/(k_B T)$ ($\Delta$ is the parameter of the trap distribution, $k_B$ is the Boltzmann constant, $T$ is the absolute temperature), $V$ is the applied voltage, and $d$ is the sample thickness.

The explanation of our experimental data, as in the case for CIS-based solar cells with $1 < m < 2$ (the transition part to...
The steady-state photocurrent spectra of all amorphous thin-film structures were measured with an applied external electric field of $E = 5 \times 10^4$ V/cm, that is, in the region where the $I$–$V$ characteristics exhibit linear behavior. In our previous papers [7,8], we have presented the experimental results of steady-state and transient photocurrents of amorphous Ge$_x$As$_y$Se$_{1-x}$ and (As$_4$S$_3$Se$_3$)$_{1-x}$Sn$_x$ thin films. It was demonstrated that the dependence of the steady-state photocurrent on the light intensity for all investigated amorphous thin films is non-linear, and can be described by the expression $I_{\text{ph}} \sim F^\alpha$, with the parameter $1.0 \leq \alpha \leq 0.5$, which indicates the existence of mono- and bimolecular recombination mechanisms in these materials.

Figure 5 and Figure 6 show the steady-state photocurrent spectra of separate amorphous thin films and HS Al–As$_{0.40}$S$_{0.30}$Se$_{0.30}$/Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$/Ge$_{0.30}$As$_{0.04}$S$_{0.66}$–Al with positive or negative voltage applied to the illuminated top Al electrode. The steady-state photocurrent spectra show that the maximum of photosensitivity of the component layers of the multilayer HS is in a good agreement with the absorption spectra. The spectral distribution of the photocurrent for the HS is situated between the distributions of the separate layers and depends on the polarity of the illuminated electrode. The position of the maximum of Al–Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$–Al, located at $h\nu = 2.47$ eV and $h\nu = 2.95$ eV, indicates a probable separation of this sample into multiple phases that contain trigonal ($\text{AsS}_3/2$) and tetrahedral ($\text{GeS}_4$) structural units, respectively. Figure 5 and Figure 6 also show that for the thin-film structure Al–As$_{0.40}$S$_{0.40}$Se$_{0.30}$–Al and for the multilayer HS the position of the maximum in the spectral distribution of the photocurrent depends on the polarity of the applied voltage. When a positive voltage is applied to the illuminated Al electrode, the maximum is shifted toward the region of higher energies of the spectrum. As was mentioned in [8], this behavior can be associated with contact phenomena at the interfaces between metallic electrode and amorphous layers, as well as with the drift and surface recombination of the non-equilibrium carriers [14].

The normalized photocurrent spectra of the multilayer thin-film HS at different, positive and negative, values of the voltage applied to the illuminated top Al electrode ($U = 1.0$–$10.0$ V) are presented in Figure 7 and Figure 8. It is evident that increasing the positive applied voltage leads to a shift of the photocurrent maximum toward higher energies (Figure 7). An increase of the negative applied voltage leads to a shift of the photocurrent maximum toward lower energies (Figure 8). This shift can be attributed to the different resistance values and the different $I$–$V$ characteristics of the component layers of the HS. The distribution of the applied voltage among the layers depends on the
photocurrent. Therefore, the contribution of each layer to the total photocurrent is different. When the applied voltage has a positive polarity (0.1–200 V), the structure of the photocurrent spectra is more complicated. It was observed that at low applied voltages (0.1–0.5 V) the photocurrent is negative, and at $U = 1.0$ V the photocurrent changes to positive values. This phenomenon can occur when the external applied field becomes higher than the internal electrical field. For amorphous semiconductors the barrier height of an Al–semiconductor contact, with a work function of $\phi_m = 4.18$ eV, is $\phi_b = 0.40–0.75$ eV [14,17]. This is very important from a practical point of view, because there is the possibility to build photodetectors with different signs of the photocurrent. This phenomenon of photocurrent sign inversion through polarization was also observed for photodetectors based on ZnAs$_2$ anisotropic crystals [18].

Figure 8 shows that, besides the main peaks in the spectral distribution of the photocurrent at 2.56–2.92 eV, there are also significant peaks in the region of 1.26–2.11 eV. When the voltage applied to the top Al electrode has a positive polarity, the maximum at 2.82–2.92 eV is attributed to the absorption of light in the wide bandgap of Ge$_{0.30}$As$_{0.04}$S$_{0.66}$ ($E_g \approx 3.0$ eV). It weakly depends on the applied voltage, $\Delta(h\nu) = 0.1$ eV. When the voltage applied to the top Al electrode has a negative polarity, the maximum at 2.56–2.86 eV is also attributed to the absorption of light in the wide bandgap of Ge$_{0.30}$As$_{0.04}$S$_{0.66}$ ($E_g \approx 3.0$ eV). However, it depends more strongly on the applied voltage, $\Delta(h\nu) = 0.3$ eV. The other maxima at 1.26–2.11 eV correspond to the narrow-bandgap materials Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$ and Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$. Some of them are also present in the spectral distribution of the single-layer structures (Figure 5 and Figure 6). Spectral position and amplitude of these maxima depend, among others, on the value and polarity of the applied voltage.

Figure 9 shows the position of the photocurrent maximum of the different components of the HS and of the HS itself at positive (curve 1) and at negative (curve 2) polarity of the voltage applied to the illuminated top Al electrode. The spectral position of the maximum of the photocurrent of the HS has an intermediate value with respect to the component layers.

Figure 10 shows the peak position in the spectral distribution of the photocurrent of the amorphous thin-film HS as a function of the applied voltage with positive (1) and with negative (2) polarity. At low applied voltages with positive polarity ($U \leq 10$ V), a shift of the maximum of the photocurrent to higher photon energies (from 2.3 eV to 2.9 eV) is observed. When the applied voltage is increased further up to $U = 100$ V,
the maximum returns to its initial position ($\hbar \nu = 2.3$ eV, curve 1). At negative polarity, the maximum of the photocurrent is shifted only to lower photon energies (from 2.75 to 2.2 eV).

Figure 11 shows the magnifying power $K$ (i.e., the ratio between photocurrent and dark current, $K = I_{\text{ph}}/I_{\text{dark}}$) of the thin-film structures and the HS both at positive (curve 1) and at negative (curve 2) polarity of the applied voltage. The value of $K$ is higher at positive polarity than at negative polarity. Also, the amorphous Al–As$_{0.40}$S$_{0.30}$Se$_{0.30}$–Al thin-film structure shows the highest values of $K$.

Figure 12 shows the dependence of the magnifying power $K$ of the amorphous thin-film HS as a function of the applied voltage with positive (1) and negative (2) polarity. The photosensitivity is higher when the voltage applied at the illuminated top Al electrode has a positive polarity, which was also demonstrated for structures of single thin films [7].

Conclusion

The experimental results regarding optical absorption and steady-state photoconductivity of both amorphous single-layer structures, Al–As$_{0.40}$S$_{0.30}$Se$_{0.30}$–Al, Al–Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$–Al, and Al–Ge$_{0.30}$As$_{0.04}$S$_{0.66}$–Al and the heterostructure Al–As$_{0.40}$S$_{0.30}$Se$_{0.30}$/Ge$_{0.09}$As$_{0.09}$Se$_{0.82}$/Ge$_{0.30}$As$_{0.04}$S$_{0.66}$–Al have been discussed. The spectral distribution of the photocurrent depends on the value and on the polarity of the voltage applied to the illuminated top Al electrode. It was found, that at low applied voltages with positive polarity ($U \leq 10$ V) the maximum of the photocurrent is shifted to higher photon energies (from 2.3 to 2.9 eV) with increasing voltage. When the applied voltage is further increased up to $U = 100$ V, the maximum returns to its initial position.

The shift of the photocurrent maximum may be related to the component layers of the HS. Because the resistance of the wide-bandgap material (Ge$_{0.30}$As$_{0.04}$S$_{0.66}$) is higher, at low voltages the electrical field mainly is distributed in this material, which leads to a photoconductivity maximum in the high-energy region. When the voltage is further increased, the electrical field is also distributed in the narrow-bandgap materials and the position of the photocurrent maximum is shifted in to the low-energy region. For GeTe and GeSe films, an analogous shift was explained by bulk phenomena in the material, not by contact phenomena [19]. In contrast, when the applied voltage has a negative polarity, the maximum of the photocurrent is shifted only to lower photon energies (from 2.75 to 2.2 eV).
Besides that, the magnifying power, \( K = \frac{I_{ph}}{I_{dark}} \), depends on the composition of the thin-film structures and of the HS and on the polarity of the applied voltage as well. Moreover, it was demonstrated that higher values of the magnifying power \( K \) were reached at positive polarity for all investigated amorphous thin-film structures. This result can be explained by drift processes of non-equilibrium carriers in amorphous semiconductors as well as by the contact phenomena between interfaces of different amorphous materials and the metallic electrodes.

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