A Single Residue Switch for Mg$^{2+}$-dependent Inhibition Characterizes Plant Class II Diterpene Cyclases from Primary and Secondary Metabolism*

Received for publication, March 13, 2010, and in revised form, April 21, 2010 Published, JBC Papers in Press, April 29, 2010, DOI 10.1074/jbc.M110.123307

Francis M. Mann†, Sladjana Prisic‡1, Emily K. Davenport†, Mara K. Determan†, Robert M. Coates†, and Reuben J. Peters†3

From the †Department of Biochemistry, Biophysics, and Molecular Biology, Iowa State University, Ames, Iowa 50011 and the §Department of Chemistry, University of Illinois, Urbana, Illinois 61801

Class II diterpene cyclases mediate the acid-initiated cycloisomerization reaction that serves as the committed step in biosynthesis of the large class of labdane-related diterpenoid natural products, which includes the important gibberellin plant hormones. Intriguingly, these enzymes are differentially susceptible to inhibition by their Mg$^{2+}$ cofactor, with those involved in gibberellin biosynthesis being more sensitive to such inhibition than those devoted to secondary metabolism, which presumably limits flux toward the potent gibberellin phytohormones. Such inhibition has been suggested to arise from intrasteric Mg$^{2+}$ binding to the DXDD motif that cooperatively acts as the catalytic acid, whose affinity must then be modulated in some fashion. While further investigating class II diterpene cyclase catalysis, we discovered a conserved basic residue that seems to act as a counter ion to the DXDD motif, enhancing the ability of aspartic acid to carry out the requisite energetically difficult protonation of a carbon-carbon double bond and also affecting inhibitory Mg$^{2+}$ binding. Notably, this residue is conserved as a histidine in enzymes involved in gibberellin biosynthesis and as an arginine in those dedicated to secondary metabolism. Interchanging the identity of these residues is sufficient to switch the sensitivity of the parent enzyme to inhibition by Mg$^{2+}$. These striking findings indicate that this is a single residue switch for Mg$^{2+}$ inhibition, which not only supports the importance of this biochemical regulatory mechanism in limiting gibberellin biosynthesis, but the importance of its release, presumably to enable higher flux, into secondary metabolism.

The labdane-related diterpenoids are a large superfamily of natural products, with almost 7,000 known (1). The founding members of this natural products superfamily seem to be the gibberellin (GA)$^4$ phytohormones. Given the essential role of GA for normal growth and development in higher plants, there is an absolute requirement for the corresponding enzymatic genes (2). These then form a biosynthetic reservoir from which additional, more specialized/secondary metabolism can, and clearly has, evolved (3). Accordingly, it is perhaps not surprising that the majority of the known labdane-related diterpenoids are produced by plants.

Labdane-related diterpenoid biosynthesis is uniquely initiated by sequential cyclization and/or rearrangement reactions, the first of which is catalyzed by class II diterpene cyclases, whose bicyclic product is then elaborated by class I diterpene synthases (4, 5). However, the predominant metabolic fate of the universal diterpenoid precursor (E,E,E)-geranylgeranyl diphosphate (GGPP) in plastids, where such biosynthesis is initiated in plants, is incorporation into photosynthetic pigments (i.e. carotenoids and the phylty side chain of chlorophyll) (Fig. 1). Indeed, the potent biological activity of GA presumably limits the amount of flux necessary for such phytohormone production (6, 7). Nevertheless, the more specialized metabolites that evolved from GA typically require higher flux (e.g. as antibiotic components of a defense response to microbial infection (8)).

The enzymatic structure-function relationships underlying the acid initiated cycloisomerization reaction mediated by the class II diterpene cyclases that initiate labdane-related diterpenoid biosynthesis remain unclear. The requisite carbon-carbon double bond (C=C) protonation is energetically difficult, yet appears to be catalyzed by a DXDD motif wherein the “first” and “last” aspartates act to increase the acidity of the “middle” aspartate, which then carries out such protonation. This hypothesis was originally based on the highly conserved nature of this motif and analogies to the much better understood C=C protonation reaction catalyzed by bacterial squalene-hopene cyclases (SHCs), which contain an analogous catalytic motif (9). Mutational analysis has since demonstrated the importance of the corresponding aspartates in class II diterpene cyclases (10, 11), and we have reported evidence in a preliminary communication strongly supporting similar cooperative catalytic acid function for this motif (12). Furthermore, it has recently been suggested that class II diterpene cyclases are descended from SHC (13). However, in SHC, it is known that the ability of the

---

* This work was supported by grants from the National Science Foundation (MCB0919735) and National Institutes of Health (GM076324) to R. J. P.

† Both authors contributed equally to this work.

‡ The on-line version of this article (available at http://www.jbc.org) contains supplemental Tables S1–S3, Fig. S1, and additional references.

1 Present address: Division of Infectious Diseases, Children’s Hospital/Harvard Medical School, Boston, MA 02115.

2 To whom correspondence should be addressed: Iowa State University, 4216 Molecular Biology Bldg., Ames, IA 50011. Fax: 515-294-8580; E-mail: rjpete@iastate.edu.

3 The abbreviations used are: GA, gibberellin phytohormones; SHC, squalene-hopene cyclase; AsAG, abietadiene synthase from Abies grandis; CPP, copa-
corresponding DXDD motif to carry out this energetically difficult reaction further requires the presence of a basic residue (conserved as histidine or arginine (14, 15)) to act as a counter ion (16), but this residue is not conserved in class II diterpene cyclases.

In the course of our initial characterization of the ent-copalyl diphosphate (ent-CPP) synthase (CPS) from Arabidopsis thaliana (AtCPS), we noted that this GA-associated class II diterpene cyclase suffered from striking inhibition exerted by its divalent magnesium ion (Mg$^{2+}$) cofactor, which also is synergistic with previously reported GGPP substrate inhibition, and which we hypothesized occurred via “inhibitory” Mg$^{2+}$ binding to the DXDD motif (17). Intriguingly, we further found that the class II diterpene cyclase activity of the resin acid (i.e. more specialized or secondary metabolism) dedicated abietadiene synthase from Abies grandis (AgAS) was much less susceptible to such Mg$^{2+}$-mediated inhibition, although it also contains the DXDD motif. We then hypothesized that the observed Mg$^{2+}$ inhibitory effect represented a biochemical feed-forward inhibitory mechanism acting to limit flux toward production of the potent GA phytohormones in plant plastids, with an as-yet-unidentified residue acting to determine susceptibility to such intrasteric inhibitory Mg$^{2+}$ binding to the DXDD motif (17). Here, we report results that not only provide insight into the enzymatic mechanism, but also Mg$^{2+}$-dependent regulation of class II diterpene cyclases, including how this is modified in response to the selective advantages provided by differential regulation of biosynthetically related/derived, but physiologically distinct (i.e. primary/GA versus more specialized/secondary) metabolic fluxes.

**EXPERIMENTAL PROCEDURES**

**General**—The preparation of reference samples for products and substrate (+/−)-14,15-oxido-geranylgeranyl diphosphate (oxido-GGPP) have been described previously (12, 17). GGPP was purchased from Sigma-Aldrich (St. Louis, MO) or Isoprenoids, LC (Tampa, FL). Unless otherwise noted, all other chemicals were purchased from Fisher Scientific (Loughborough, Leicestershire, UK), and molecular biology reagents from Invitrogen. Sequence alignments were carried out using the AlignX program from the Vector NTI software package (Invitrogen), using the default parameters.

**Enzymatic Analyses**—Recombinant pseudomature AtCPS and AgAS were expressed, purified, and assayed as described previously (17). Site-directed mutagenesis was carried out via PCR amplification of pENTR (Gateway, Invitrogen) clones using overlapping mutagenic primers. The resulting mutant genes were verified by complete sequencing prior to transfer via directional recombination to pDEST14 and pDEST17 expression vectors. His-tag purification was enabled by pDEST17 constructs, which encode an N-terminal His$_6$ tag, allowing purification with nickel-nitrilotriacetic acid superf low resin (EMD Chemicals, Gibbstown, NJ), used according to the manufacturer’s instructions, with the resulting enzymes being >95% pure as judged by SDS-PAGE analysis. Kinetic parameters were calculated from fitting the observed data to the substrate inhibition equation (KaleiGraph 4.0, Synergy Software; Reading, PA). For data that could not be fit to the substrate inhibition equation (Mg$^{2+}$ kinetics with AtCPS,D377A; AtCPS,D380A; AtCPS,H331A; and AgAS,D621A/R356H; as indicated by $R^2 < 0.9$), kinetic parameters were calculated by partial double reciprocal plots (i.e. using the increasing rate data at lower Mg$^{2+}$ concentrations to calculate $K_m$ and the subsequently decreasing rate data at higher Mg$^{2+}$ concentrations to calculate $K_i$, with the resulting kinetic constants marked with an asterisk in Table 1 and supplemental Table S3), with the exception of AtCPS,H331R, for which there was no detectable substrate inhibition and was fit to the Michaelis-Menten equation ($R^2 = 0.97$). For clarity and consistency, all kinetic plots are shown with a smooth fit line simply connecting the observed data points.

**RESULTS**

To further investigate the mechanism by which the DXDD motif initiates class II diterpene cyclization, we carried out additional mutational analysis using alanine, glutamate, and asparagine substitutions for each of these aspartates in AtCPS. Appreciable mutant activity was only found with substitution of glutamate for the first, or asparagine for the last, aspartates (supplemental Table S1). These results are consistent with previous mutational analysis of this motif in AgAS (10), and previous reports of CPS that contain glutamate in place of the first aspartate (18, 19). Together, this suggests the possibility of a polar interaction wherein the first asparate carries a negative charge that is presumably stabilized by interaction with the last asparate.

To identify a catalytic basic residue that might act as a counter ion to the DXDD motif in class II diterpene cyclases, the domains that contain the class II active site (20) were examined

**FIGURE 1. Various metabolic fates of GGPP in plant plastids.** The production of photosynthetic pigments and GA is conserved in all higher plants, whereas the production of resin acids initiated by the AgAS studied here is shown as an example of more specialized/secondary labdane-related diterpenoid metabolism. OPP, diphosphate ester group.
for similarly conserved (His/Arg) positions. Only a single such position was found (supplemental Fig. S1), and an alanine substitution for the corresponding His in AtCPS (H331A) led to a significant reduction in enzymatic activity.

Expanding upon our initial study of the catalytic aspartate residues Asp379 and Asp380 (12), we examined the kinetic parameters of the D377A and H331A mutants (Table 1). Each mutant exhibited a significant reduction in activity (>1,000-fold decrease in $k_{cat}$), but neither appeared to drastically affect substrate binding (>5-fold increase in $K_m$). In combination with our previous study, these results are consistent with the hypothesis that the DXDD motif is not involved in productive GGPP binding. However, the previously noted GGPP substrate inhibition effect was slightly more pronounced in the H331A mutant, although no such substrate inhibition was seen with the D379A or D380A mutants, indicating a drastic increase in $K_i$ for GGPP.

Additional kinetic studies were carried out with the GGPP analog (+/−)-14,15-oxidogeranylgeranyl diphosphate (oxido-GGPP). This substrate is more easily protonated, due to the epoxide ring substitution for the C14−C15 C=C of GGPP, and its cyclization yields 3-hydroxycopalyl diphosphate epimers (Fig. 2). We have characterized previously AtCPS,D379A and AtCPS,D380A with this substrate analog (12), and here, we report analogous investigations with AtCPS,D377A and AtCPS,H331A. As previously reported, AtCPS,D379A is essentially unable to cyclize either GGPP or oxido-GGPP, whereas AtCPS,D380A is significantly more active with oxido-GGPP than GGPP, as were the AtCPS,D377A and AtCPS,H331A reported here (Table 2). Together, these results support a model of class II diterpene cyclase activity with a catalytic tetrad composed of the DXDD motif and a basic (His/Arg) residue counter ion, much as found in SHC.

We have previously proposed that the DXDD motif acts as an inhibitory Mg$^{2+}$ binding site (17) and here investigated the effects of alanine substitution for the catalytic tetrad on the response of AtCPS activity to varying concentrations of Mg$^{2+}$ (Fig. 3). Both the D377A and D380A mutants exhibited reduced Mg$^{2+}$ inhibition relative to wild-type AtCPS, with considerably less drastic decreases in activity above the optimal concentration of Mg$^{2+}$, indicating that at least the first and last aspartates of the DXDD motif are involved in inhibitory Mg$^{2+}$ binding. (Unfortunately, AtCPS,D379A did not retain sufficient activity for such analysis.) In addition, AtCPS,D379A exhibited positive cooperativity, indicating complex interactions between multiple activating Mg$^{2+}$. By contrast, the H331A mutant exhibited accentuated Mg$^{2+}$ inhibition. While of the opposite effect, this result also indicates proximity to the DXDD motif and further supports the role of this residue as a catalytic counter ion to the DXDD motif. Furthermore, this indicates that His$^{331}$ acts

### Table 1

| AtCPS$^a$ | $k_{cat}$ $s^{-1}$ | $K_m$ μM | $V_{max}$ μM/s |
|-----------|------------------|---------|----------------|
| Wild-type | 3                | 3       | 5              |
| H331A     | 2 × 10$^{-4}$    | 2       | 3              |
| D377A     | 2 × 10$^{-4}$    | 2       | 6              |
| D379A     | 3 × 10$^{-7}$    | 16$^b$  | 5              |
| D380A     | 7 × 10$^{-4}$    | 5       |                |

$^a$ Assays contained the 0.1 mM MgCl$_2$ required for optimal activity with the wild type enzyme.

$^b$ Kinetic constants derived from partial double reciprocal plot fits, as described under “Experimental Procedures.”

### Table 2

| AtCPS$^a$ | $k_{cat}$GGPP $s^{-1}$ | $k_{cat}$oxido-GGPP $s^{-1}$ | $V_{max}$GGPP μM/s | $V_{max}$oxido-GGPP μM/s | $V_{max}$/GGPP $\mu M/s$ |
|-----------|------------------------|-----------------------------|---------------------|--------------------------|--------------------------|
| Wild-type | 1.2 ± 0.1              | 0.7 ± 0.1                   | 0.6                 |                          |                          |
| H331A     | (5 ± 1) × 10$^{-5}$    | (7 ± 1) × 10$^{-4}$         | 14                  |                          |                          |
| D377A     | (6 ± 1) × 10$^{-5}$    | (2 ± 1) × 10$^{-3}$         | 33                  |                          |                          |
| D379A     | <10$^{-9}$             | <10$^{-9}$                  | <10                 |                          |                          |
| D380A     | (4 ± 2) × 10$^{-3}$    | (2 ± 1) × 10$^{-2}$         | 50                  |                          |                          |

$^a$ Error represents standard error from two independent measurements.
antagonistically to reduce the affinity of the DXDD motif for inhibitory Mg$^{2+}$ binding, presumably due to electrostatic repulsion between this divalent metal cation and the proposed catalytically relevant positively charged form of the histidine.

Intriguingly, we noted that the catalytic basic residue is conserved as a His in class II diterpene cyclases with a demonstrated role in GA biosynthesis, whereas those involved in secondary/specialized metabolism exclusively contained an Arg at this position (Fig. 4). On the basis of this striking conservation pattern, we hypothesized that this residue might control the differential susceptibility of class II diterpene cyclases to inhibition by Mg$^{2+}$. This was investigated by making reciprocal mutations in AtCPS, which is involved in GA (i.e. primary) metabolism and exhibits strong synergistic GGPP and Mg$^{2+}$ inhibition, and AgAS, which is a bifunctional diterpene synthase dedicated to resin acid (i.e. secondary) metabolism and is much less susceptible to this substrate/cofactor inhibition effect (17). Specifically, we constructed a set of “switch” mutants at the conserved His/Arg site, AtCPS:H331R and AgAS:D621A/R356H. (AgAS, D621A has been characterized previously as lacking class I activity without any effect on class II activity (11), allowing examination of the effect of mutations on class II activity in isolation.)

Both mutants were active but were not well expressed. Addition of a His$_6$ tag improved the yields and allowed for rapid purification. Kinetic parameters were obtained for these constructs and compared with equivalently tagged parental constructs (i.e. wild-type AtCPS and AgAS,D621A). The tagged parental constructs exhibited $<$10-fold loss of catalytic activity (i.e. $k_{cat}$) and retained $K_m$ and $K_i$ for GGPP equivalent to that of the untagged enzymes (supplemental Table S2). The effects of exchanging the His/Arg residues on sensitivity to inhibitory Mg$^{2+}$ binding were obtained by assaying each enzyme across a range of Mg$^{2+}$ concentrations. Critically, these exchange mutants retain significant catalytic activity, with a $<$10-fold effect on $k_{cat}$. Nevertheless, both exhibited striking alteration of their susceptibility to inhibition by Mg$^{2+}$ relative to the parental wild-type enzymes (Fig. 5). In particular, whereas the AtCPS wild-type enzyme has a $K_i$ for Mg$^{2+}$ of 0.7 mM, AtCPS:H331R is no longer appreciably inhibited by Mg$^{2+}$ in the range of concentrations tested here. Further, whereas AgAS:D621A has a $K_i$ for Mg$^{2+}$ of 12 mM, AgAS:D621A/R356H has a dramatically increased $K_i$ of 0.25 mM (supplemental Table S3).

**DISCUSSION**

Class II diterpene cyclases catalyze the committed step in biosynthesis of the large class of labdane-related diterpenoid natural products, which range from the gibberellin (GA) phytohormones to more specialized/secondary metabolites typically utilized in defense (3). In plants, these enzymes are localized in plastids (21) and divert GGPP away from the production of phytosynthetic pigments (i.e. carotenoids and the phytyl side chain of chlorophyll) (22). Thus, understanding the mechanism and regulation of these important, gate-keeping enzymes can provide key insights into understanding how plants mediate and control metabolic flux within their plastids.

Of particular relevance to the studies discussed here, GA is only required in very small amounts *in planta*, whereas secondary metabolites can be produced in very large quantities. Because *A. thaliana* only produces GA and no other labdane-related diterpenoid natural products, AtCPS is dedicated to GA biosynthesis. Furthermore, AtCPS is found throughout fast growing tissues, as well as in the vasculature (23). Such widespread enzymatic distribution indicates that relatively limited activity would be sufficient to mediate the small amounts of metabolic flux necessary for GA biosynthesis. By contrast, conifer resin is found in anatomically specialized structures, such as the resin blisters in *Abies grandis* saplings. It has been shown...
that the relevant diterpene synthases, such as AgAS, are localized in the surrounding epithelial cells (24), which are then presumably responsible for production of all the observed resin, necessitating an enormous flux into such biosynthesis (i.e. these cells produce several times their own volume of resin). Thus, these disparate (i.e. primary/GA versus secondary) metabolic processes presumably require distinct regulatory mechanisms.

Although no structural information has yet been reported for class II diterpene cyclases, it has been suggested that these are descended from the triterpene SHCs, which catalyze mechanistically similar reactions (13). Based on analogies to the better understood SHC, it has been suggested that the class II diterpene cyclases utilize the DXDD motif they share in common with SHC, to act as the catalytic acid in C=C protonation-initiated cyclization (10, 11). In particular, that the first and last aspartates carry a negative charge that activates the middle aspartate to carry out the requisite energetically difficult C=C protonation (12), a hypothesis consistent with the data reported here.

In addition to the DXDD motif, SHC further require a catalytic basic residue, conserved as a His or Arg (14, 15) that seems to act as a counter ion to the negative charge carried by the first and last aspartates and which directly interacts with the middle aspartate (16). Although this residue is not conserved in class II diterpene cyclases, we also have identified here a catalytic basic residue, similarly conserved as His or Arg, in plant class II diterpene cyclases that seems to have an analogous function, serving to increase the ability of the middle aspartate to carry out C=C protonation. While clearly not phylogenetically conserved (i.e. in plant class II diterpene cyclases this catalytic counter ion precedes, while that in SHC follows, the DXDD motif) these mechanistically similar enzymes then both utilize a DXDD+His/Arg tetrad that acts cooperatively as the catalytic acid, providing significant insight into the enzymatic mechanism utilized by class II diterpene cyclases.

Class II diterpene cyclases require the presence of Mg$^{2+}$ for full catalytic activity (10, 17), presumably for binding of the diphosphate moiety of their GGPP substrate, whereas divalent metal ions are not required for binding of the squalene olefin substrate of SHC (25). Intriguingly, Mg$^{2+}$ exhibits a dramatic biphasic effect on the activity of the AtCPS involved in GA biosynthesis, with a very rapid decrease in activity above the optimal concentration, and this effect is much less pronounced with class II enzymes involved in more specialized/secondary metabolism (i.e. AgAS) (17). Furthermore, Mg$^{2+}$ levels vary in plant plastids where class II diterpene cyclases are found, specifically in response to light (26), over a similar sub- to low millimolar range as that observed to affect class II enzymatic activity. Thus, we have previously hypothesized that the observed selective Mg$^{2+}$-dependent suppression may act as a physiologically relevant feed-forward inhibition mechanism limiting flux toward the potent GA hormones, with inhibitory Mg$^{2+}$ binding to the catalytic DXDD motif, whereas an unidentified residue acts as a regulatory switch that modulates this interaction (17).

This hypothesis was further bolstered by the investigation of enzymatic activity presented here, which also indicated an interaction between such inhibitory Mg$^{2+}$ binding and the catalytic counter ion basic residue. In addition to being consistent with the postulated direct interaction of this residue with the DXDD motif, the enhanced inhibitory Mg$^{2+}$ binding observed upon removal of the basic catalytic counter ion suggested the possibility that this residue might be the postulated regulatory switch.

Close examination of the alignment of class II enzymes focusing on those with known physiological roles revealed a striking conservation pattern wherein this position is conserved as a His in those CPS with a demonstrated role in GA biosynthesis and as an Arg in those class II enzymes functioning in more specialized/secondary metabolism (Fig. 4). Hypothesizing that this might represent the previously postulated regulatory switch, we reciprocally exchanged the corresponding catalytic basic residues in AtCPS and AgAS and found that this led to dramatically different sensitivity to inhibition by Mg$^{2+}$ in both cases (Fig. 5). These results strongly indicate that the identity of the residue at this catalytic basic counter ion position represents the previously hypothesized regulatory switch for plant class II diterpene cyclases, consistent with the conservation pattern noted above. We hypothesize that the positively charged form of this catalytic basic residue repels Mg$^{2+}$, with the lower pK$_a$ of His relative to Arg providing a higher probability that this side chain will be in the unprotonated/neutral

---

**FIGURE 6. Proposed model for the DXDD+His/Arg catalytic tetrad, with speculative hydrogen bonding interactions among the catalytic residues and the natural GGPP substrate.** A, productive binding in the active site of CPS involved in GA biosynthesis. B, inhibitory binding of Mg$^{2+}$ preventing GGPP cyclization in the active site of CPS involved in GA biosynthesis. C, productive binding of GGPP in the class II active site of diterpene synthases involved in more specialized/secondary metabolism. The catalytic aspartates are labeled by their relative position within the DXDD motif.
form, allowing inhibitory Mg\(^{2+}\) binding to the DXDD motif (Fig. 6), although steric effects also are possible.

Regardless of the exact mechanism, the striking conservation pattern and biochemical effect revealed here strongly support physiological relevance for the previously hypothesized synergistic feed-forward inhibition exerted by GGPP and Mg\(^{2+}\) on the CPS involved in GA biosynthesis to limit flux toward this potent phytormone. In addition, these results further support the importance of removing such limitation to enable higher flux in the context of more specialized/secondary metabolism, where rapid production of relatively large amounts of (e.g. antibiotic) natural products would provide obvious advantages for such a defensive response. Accordingly, our results provide insight into the enzymatic mechanism of plant class II diterpene cyclases and also support the relevance of an intertwined biochemical regulatory mechanism, which appears to operate as a single residue switch that is flipped in response to the selective advantages provided by differential regulation of biosynthetically related/derived but physiologically distinct (i.e. primary/GA versus more specialized/secondary) metabolic fluxes.

REFERENCES

1. Buckingham, J. (2002) Dictionary of Natural Products, www.chemnetbase.com
2. Fleet, C. M., and Sun, T. P. (2005) Cur. Opin. Plant Biol. 8, 77–85
3. Peters, R. J. (2006) Phytochemistry 67, 2307–2317
4. Christianson, D. W. (2006) Chem. Rev. 106, 3412–3442
5. Wendt, K. U., and Schulz, G. E. (1998) Structure 6, 127–133
6. Alabadi, D., Gil, J., Blázquez, M. A., and García-Martínez, J. L. (2004) Plant Physiol. 134, 1050–1057
7. García-Martínez, J. L., and Gil, J. (2001) J. Plant Growth Regul. 20, 354–368
8. Lloyd, J. C., and Zakhleniuk, O. V. (2004) J. Exp. Bot. 55, 1221–1230
9. Bohlmann, J., Meyer-Gauen, G., and Croteau, R. (1998) Proc. Natl. Acad. Sci. U.S.A. 95, 4126–4133
10. Peters, R. J., and Croteau, R. B. (2002) Biochemistry 41, 1836–1842
11. Peters, R. J., Ravn, M. M., Coates, R. M., and Croteau, R. B. (2001) J. Am. Chem. Soc. 123, 8974–8978
12. Prisic, S., Xu, J., Coates, R. M., and Peters, R. J. (2007) ChemBioChem 8, 869–874
13. Cao, R., Zhang, Y., Mann, F. M., Huang, C., Mukkamala, D., Hudock, M. P., Mead, M., Prisic, S., Wang, K., Lin, F. Y., Chang, T. K., Peters, R. J., and Oldfield, E. (2010) Proteins, in press
14. Sato, T., and Hoshino, T. (1999) Biosci. Biotechnol. Biochem. 63, 2189–2198
15. Merkofer, T., Pale-Grosdemange, C., Wendt, K. U., Rohmer, M., and Porralla, K. C. (1999) Tetrahedron Lett. 40, 2121–2124
16. Wendt, K. U. (2005) Angew. Chem. Int. Ed. Engl. 44, 3966–3971
17. Prisic, S., and Peters, R. J. (2007) Plant Physiol. 144, 445–454
18. Prisic, S., Xu, M., Wilderman, P. R., and Peters, R. J. (2004) Plant Physiol. 136, 4228–4236
19. Harris, L. J., Saparno, A., Johnston, A., Prisic, S., Xu, M., Allard, S., Kathiresan, A., Ouellet, T., and Peters, R. J. (2005) Plant Mol. Biol. 59, 881–894
20. Peters, R. J., Carter, O. A., Zhang, Y., Matthews, B. W., and Croteau, R. B. (2003) Biochemistry 42, 2700–2707
21. Sun, T. P., and Kamiya, Y. (1994) Plant Cell 6, 1509–1518
22. McGarvey, D. J., and Croteau, R. (1995) Plant Cell 7, 1015–1026
23. Silverstone, A. L., Chang, C., Krol, E., and Sun, T. P. (1997) Plant J. 12, 9–19
24. Keeling, C. I., and Bohlmann, J. (2006) Phytochemistry 67, 2415–2423
25. Wendt, K. U., Schulz, G. E., Corey, E. J., and Liu, D. R. (2000) Angew. Chem. Int. Ed. Engl. 39, 2812–2833
26. Ishijima, S., Uchibori, A., Takagi, H., Maki, R., and Ohnishi, M. (2003) Arch. Biochem. Biophys. 412, 126–132
27. Ait-Ali, T., Swain, S. M., Reid, J. B., Sun, T., and Kamiya, Y. (1997) Plant J. 11, 443–454
28. Bensen, R. J., Johal, G. S., Crane, V. C., Tossberg, J. T., Schnable, P. S., Meeley, R. B., and Briggs, S. P. (1995) Plant Cell 7, 75–84
29. Sakamoto, T., Miura, K., Itoh, H., Tatsumi, T., Ueguchi-Tanaka, M., Ishiyama, K., Kobayashi, M., Agrawal, G. K., Takeda, S., Abe, K., Miyao, A., Hirochika, H., Kitano, H., Ashikari, M., and Matsuoka, M. (2004) Plant Physiol. 134, 1642–1653
30. Otomo, K., Kenmoku, H., Oikawa, H., König, W. A., Toshima, H., Mitsuhashi, W., Yamane, H., Sassa, T., and Toyomasu, T. (2004) Plant J. 39, 886–893
31. Vogel, B. S., Wildung, M. R., Vogel, G., and Croteau, R. (1996) J. Biol. Chem. 271, 23262–23268
32. Schepmann, H. G., Pang, J., and Matsuda, S. P. (2001) Arch. Biochem. Biophys. 392, 263–269