Ferromagnetism of Fe$_3$Sn and Alloys

Brian C. Sales, Bayrammurad Saparov, Michael A. McGuire, David J. Singh & David S. Parker

Materials Science and Technology Division, Oak Ridge National Laboratory.

Hexagonal Fe$_3$Sn has many of the desirable properties for a new permanent magnet phase with a Curie temperature of 725 K, a saturation moment of 1.18 MA/m, and anisotropy energy, $K_1$, of 1.8 MJ/m$^3$. However, contrary to earlier experimental reports, we found both experimentally and theoretically that the easy magnetic axis lies in the hexagonal plane, which is undesirable for a permanent magnet material. One possibility for changing the easy axis direction is through alloying. We used first principles calculations to investigate the effect of elemental substitutions. The calculations showed that substitution on the Sn site has the potential to switch the easy axis direction. However, transition metal substitutions with Co or Mn do not have this effect. We attempted synthesis of a number of these alloys and found results in accord with the theoretical predictions for those that were formed. However, the alloys that could be readily made all showed an in-plane easy axis. The electronic structure of Fe$_3$Sn is reported, as are some are magnetic and structural properties for the Fe$_3$Sn$_2$, and Fe$_5$Sn$_3$ compounds, which could be prepared as mm-sized single crystals.

Strong permanent magnets are important for many clean energy applications such as wind turbines and motors for hybrid and electric vehicles. The increasing importance of permanent magnets in modern society has resulted in renewed interest in the design or discovery of new permanent magnet materials that are cheaper and use less rare earths. Given the tremendous number of potential permanent magnet materials that have been examined in the past, this is a daunting task$^{1-3}$. To be a good permanent magnet a ferromagnetic compound should have a Curie temperature, $T_C$, above 400 K, a saturation moment, $M_s$, in the range of 1 MA/m, an anisotropy energy, $K_1$, of about 4 MJ/m$^3$, and additionally a uniaxial magnetic anisotropy. For a new compound or alloy, these values can be determined or estimated from single crystal or field oriented powder magnetization data. As an example, one of the best magnetic materials, Nd$_2$Fe$_{14}$B, has $T_C = 588$ K, $M_s = 1.28$ MA/m, $K_1 = 4.9$ MJ/m$^3$ and an easy magnetic axis along the tetragonal c axis$^4$. While the values of $T_C$, $M_s$, and $K_1$ determine the ultimate limits for the performance of a new permanent magnet, they are necessary but not sufficient. Besides the intrinsic bulk properties it is necessary to have a material in which the microstructure can be controlled to yield high coercivity; this is limited by $K_1$ but depends in a complicated way on the microstructure at the mesoscale. Therefore, while good intrinsic properties are essential, one also seeks a ferromagnet that is readily synthesized and amenable to processing. While there are several non-rare-earth based permanent magnets with excellent properties, such as FePt or CoPt$_3$, cost prohibits the use of these magnets in commercial applications.

Here, we report magnetic properties of iron rich Fe-Sn binary compounds and related alloys. This system was selected for study because of the relatively low cost of the starting elements and the fact that while Fe$_3$Sn has been reported to be ferromagnetic with properties consistent with a good permanent magnet material, there is little available data. We also note that the combination of a high fraction of Fe, the uniaxial crystal structure, and the presence of Sn, which while lighter than the rare-earth elements, could potentially provide a sizable contribution to the anisotropy ($p$-electrons generally have stronger spin-orbit interactions than $f$-electrons and could in a Sn compound potentially hybridize substantially with Fe-$d$ states). As mentioned, the compound Fe$_3$Sn was of particular interest because of a reported Curie temperature of 743 K and a saturation moment of 2.27 $\mu_B$ per Fe at 77 K ($M_s \approx 1.13$ MA/m)$^{5,6}$. In addition, Trumpy and co-workers$^5$ reported Mossbauer data from a polycrystalline sample indicating a uniaxial easy axis. However, as discussed below, both our first principles predictions and our direct experimental measurements show that the easy axis is in the ab-plane. Most of the work reported in this article is focused on chemical substitutions (many suggested by theoretical calculations) in Fe$_3$Sn that might flip the easy axis to uniaxial.

There are five intermetallic compounds reported in the Fe-Sn binary system, specifically, FeSn$_2$, FeSn, Fe$_3$Sn$_2$, Fe$_5$Sn$_3$, and Fe$_3$Sn. A discussion of the crystal structures, phase diagram and basic properties of the five compounds is given by Giefers and Nicol$^7$. For potential permanent magnets we focused on only the most iron rich phases: Fe$_3$Sn$_2$, Fe$_5$Sn$_3$, and Fe$_3$Sn. The Fe$_3$Sn phase was the most interesting as a potential permanent magnet.
phase, but we collected some data the Fe₃Sn₂ and Fe₅Sn₃ compounds since these latter two phases could be prepared as mm-sized single crystals

**Theoretical Predictions**

We performed first principles density functional theory (DFT) calculations with two main goals: firstly, to help understand the magnetic characteristics of the line compounds such as Fe₅Sn and Fe₃Sn₂, and secondly to provide insight as to what substitutions might improve the magnetic properties of these materials. These calculations were done using the generalized gradient approximation of Perdew, Burke and Ernzerhof (PBE-GGA). We used the general potential linearized augmented planewave (LAPW) method as implemented in the WIEN2K code. Efforts were focused on Fe₃Sn as this compound appeared to be the most promising of the compounds based on its comparatively large saturation magnetization.

We obtained a calculated saturation magnetic moment of Fe₃Sn of 1.50 Tesla, or 1.19 MA/m, in good agreement with the experimentally observed 5 K value of 1.23 MA/m. We found that the magnetic easy axis lies in the basal plane. This in-plane direction contradicts the prior report, based on inference from Mossbauer measurements, but is confirmed by our experimental measurements. The first anisotropy constant K₁ was computed. Note that for a hexagonal system such as Fe₃Sn the first term in the MAE expansion is $K_1 \sin^2(\theta)$, with \( \theta \) the angle between the magnetization and crystallographic c-axis. Physically K₁ represents the relative difficulty, or associated energy penalty, of changing the orientation of the magnetic moments under the application of a magnetic field, and is an indispensable component of attaining coercivity in a permanent magnet. K₁ was calculated as 1.59 MJ/m³, in good agreement with the value of 1.8 MJ/m³ estimated from the experimental data.

The positive sign of K₁ means that, as observed experimentally, the material is easy-planar, which is detrimental for permanent magnet applications as planar anisotropy is much less desirable than axial anisotropy. However, the magnitude of K₁ is significant – it is relatively near the 2 MJ/m³ value that Coey (Ref. 3) has cited as an enabling value for high performance permanent magnets. Indeed, if the sign of K₁ were opposite, one might, with optimization of the electron count, since varying electron count changes degeneracy and can help stabilize or destabilize orbital moment. We thus began with virtual crystal calculations for electron deficient and rich (Mn and Sb) substitution on the Sn site.

Theoretical calculations were done, with this in mind, in order to find a possible substitutional alloy based on Fe₃Sn but with uniaxial rather than planar anisotropy. A key ingredient in the anisotropy is the electron count, since varying electron count changes degeneracy and can help stabilize or destabilize orbital moment. We thus began with virtual crystal calculations for electron deficient and rich (Mn and Sb, respectively) alloying on the Fe site and also electron rich (Sb) substitution on the Sn site.

Results of these calculations are given in Table 1. As seen, for both 25% Antimony substitution, as well as a co-doping with 25% Antimony and 25% Cobalt, the sign of K₁ changes yielding the desired uniaxial anisotropy. The uniaxial anisotropy increases for increased Sb levels. While the computed magnitudes of K₁ are not as large, at least for the calculated doping levels, there may be other combinations of chemical substitutions that produce larger values of K₁, for the correct sign. The key is an increased electron count on the.Sn site. Transition metal substitutions were not found to be effective in changing the anisotropy direction. As described below, our initial synthesis efforts using these substitutions were unsuccessful and ultimately destabilized the Fe₃Sn structure. This result, however, does not necessarily mean that all such doping attempts must fail, but rather that more intensive experimental efforts along these lines are warranted.

![Figure 1](image_url) | The calculated densities-of-states of Fe₅Sn in the non-magnetic state (a) and in the ferromagnetic state (b).

### Table 1 | Calculated values of anisotropy energy, K₁, and easy magnetization direction, for various substitutions into the Fe₃Sn phase based on virtual crystal calculations. A negative value for K₁ indicates an easy magnetization axis along c

| Dopant and Concentration | Predicted Anisotropy | K₁(MJ/m³) |
|--------------------------|----------------------|-----------|
| Pure Fe₃Sn               | Planar               | 1.59      |
| 25% Mn                   | Planar               | 0.78      |
| 25% Co                   | Planar               | 0.74      |
| 25% Sb                   | Uniaxial             | -0.5      |
| 25%Co + 25%Sb            | Uniaxial             | -0.32     |
| 25%Mn + 25%Sb            | Planar               | 1.55      |

Electronic structure calculations of Fe₅Sn have not been previously reported. In Figure 1 we present the calculated electronic density of states of Fe₅Sn in the non-magnetic and ferromagnetic states. The plots show the expected characteristics of a Stoner ferromagnet, similar to Fe, specifically a high Fe d density of states at the Fermi level without moment formation, which is then strongly reduced by the exchange splitting in the ferromagnetic phase. The non-spin-polarized value is $N(E_F) = 29.05$ states/eV-unit cell, or 4.8 states/eV on a per Fe basis. The Stoner criterion $(Nf > 1$, with $N = N(E_F)/2 = 2.4$ eV⁻¹ to place it on a per spin basis, and $I = 1$ eV for 3d transition elements) is clearly satisfied. The calculated exchange splitting is ~2.5 eV, based on the spin-polarized density of states. Thus the basic physics of the moment formation is similar to the well-established understanding of bcc Fe.

As virtually all the DOS character near $E_F$ is from Fe d-states, it is instructive to compare certain of the results to those computed for elemental Fe. In particular, the calculated local moment on Fe in Fe₅Sn, at 2.41 $\mu_B$, is significantly greater than the corresponding value for Fe (2.25 $\mu_B$). In addition, the calculated total (i.e. spin-up + spin-down) magnetic DOS for Fe₅Sn is 0.78 states/eV-Fe atom.
Figure 2 | Powder x-ray diffraction patterns from (a) Fe₃Sn and (b) an Fe₅Sn₃ sample that was quenched from 827°C. The vertical lines in (b) correspond to superlattice reflections. Note that for the pure Fe₃Sn phase the (200) and (002) reflections have similar intensity in an unoriented powder.

Table 2 | Crystallographic properties of several Fe-Sn based alloys. The quench temperatures for the Fe₅Sn₃ compounds are shown in parenthesis. For the Fe₅Sn₃ crystal, crystallographic data for the Ni₂In-type subcell is shown. See Supplemental Material for details of the orthorhombic structure.

| Composition                  | a (Å)  | c (Å)  | c/a   | Cell Volume (Å³) |
|------------------------------|--------|--------|-------|------------------|
| Fe₃Sn                        | 5.4624 | 4.3544 | 0.7972| 112.52           |
| Fe₂.₅Mn₀.₅Sn₀.₉₅Sb₀.₀₅        | 5.4795 | 4.3669 | 0.7969| 113.55           |
| Fe₂.₅Mn₀.₅Sn                  | 5.4763 | 4.3636 | 0.7868| 113.33           |
| Fe₁.₅Mn₁.₅Sn₀.₉Sb₀.₁           | 5.5310 | 4.4175 | 0.7987| 117.03           |
| Fe₁.₅Mn₁.₅Sn₀.₈₅Sb₀.₁₅        | 5.5325 | 4.4192 | 0.7988| 117.17           |
| Fe₅Sn₃ (1100 K)               | 4.2187 | 5.2464 | 1.2436| 80.86            |
| Fe₅Sn₃ (1155 K)               | 4.2247 | 5.2355 | 1.2393| 80.925           |
| *Fe₅Sn₃ Crystal (hexagonal Ni₂In-type subcell) T = 173 K | 4.2063 | 5.242  | 1.2464| 80.34            |

Figure 3 | Powder x-ray diffraction data from magnetic-field-oriented Fe₃Sn powder dispersed in a clear epoxy. A field of 4 Tesla is applied along the gelcap axis until the epoxy hardened. The x-ray data unambiguously identifies the easy magnetic axis in the basal plane, which is detrimental for a potential permanent magnet, and in contradiction to the conclusion of Trumpy et al.6. based on a less direct method (Mossbauer data).

Figure 4 | Magnetization versus applied field for oriented Fe₃Sn powder, and a polycrystalline Fe₃Sn sample. Using the x-ray density of Fe₃Sn, the room temperature saturation moment is 1.18 MA/m, or about 2.37 μB/Fe. Noting that the magnetization curves merged together near 3 Tesla, indicates a value for K₁ of about 1.8 MJ/m.
(note that the unit cell contains 6 Fe), while the corresponding value for pure Fe is 1.11 states/eV-Fe atom. These facts together suggest that the moment formation in Fe₃Sn is in some sense stronger than in Fe although the Curie temperature is lower than that of Fe. This is consistent with reduced effective coordination in the Fe₃Sn structure, which in general gives narrower bands, and thus higher moments, and at the same time reduced effective exchange couplings and therefore Curie temperatures. In bcc Fe, there are eight nearest neighbors at 2.48 Å distance, while in Fe₃Sn each of the six equivalent Fe atoms has two Fe nearest neighbors at 2.49 Å and four more Fe at 2.61 Å. Note also that the nearest Sn atom is at a distance of 2.73 Å. This reduced effective coordination in a uniaxial structure also favors orbital moment formation and anisotropy on the Fe site. Our calculated Fe orbital moment is 0.073 μₘ per Fe for Fe₃Sn.

Returning to the role of Sn, as mentioned p-electrons in principle can yield stronger effects than d- or f-electrons provided that a moment is induced in the p-electron atom. In the present case, the states near E_f are almost entirely derived from Fe d-electrons. Magnetocrystalline anisotropy comes from the spin orbit interaction. We repeated the calculations of the anisotropy for Fe₃Sn with spin-orbit removed on the Sn site (but left intact on the Fe site). We find that in spite of the fact that almost all the magnetization comes from Fe, Sn does contribute a non-negligible amount to the anisotropy. The value of K₁ is reduced by ~15% to 1.36 MJ/m³ when spin-orbit is removed on the Sn site.

**Experimental Results**

A portion of the powder x-ray diffraction pattern from a polycrystalline Fe₃Sn sample and a Fe₅Sn₃ sample are shown in Figure 2. The Fe₃Sn phase is at least 95% single phase, and the crystal structure can be described by the hexagonal Ni₃Sn structure with the lattice constants given in Table 2. To assess the magnitude of the magnetic anisotropy energy, and the easy axis of magnetization for Fe₃Sn, powder of Fe₃Sn was dispersed in a clear non-magnetic epoxy inside a gelcap and exposed to a 4 Tesla magnetic field until the epoxy had hardened. A powder x-ray diffraction pattern from the oriented powder confirmed the theoretical prediction that the magnetic easy axis is in the basal plane and not along the c axis as concluded by Trumpy based on Mossbauer data. To further confirm this conclusion the two-rotation method was used to orient Fe₃Sn powder so that the magnetically hard axis lay along the cylindrical axis of the gelcap. As expected, powder x-ray diffraction data (Figure 3) showed that the hard axis is along c (i.e. the easy axis is in the basal-plane).

A rough estimate of the anisotropy energy, K₁, can be obtained from a comparison of magnetization curves from Fe₃Sn powder oriented along the easy and hard directions. These data are shown...
in Figure 4. Using the approximation that \( K_1 \approx H_u M_u / 2 \) with \( H_u \approx 3 \) Tesla (the field at which the magnetization curves along the easy and hard directions merge), \( M_u = 1.18 \) MA/m yields 1.8 MJ/m\(^3\). The experimental magnitude of \( K_1 \) is reasonably large but has the wrong sign (easy axis in plane rather than along c). Magnetization data versus temperature (300-2 K) from polycrystalline Fe\(_3\)Sn in various applied magnetic fields (not shown) gave no indication of a change of the easy axis with temperature. This experimental estimate of \( K_1 \) is remarkably close to the theoretical prediction of 1.59 MJ/m\(^3\) given above.

A planar easy axis was also found for the Fe\(_5\)Sn\(_3\), and Fe\(_3\)Sn\(_2\) phases using single crystals (Figure 5), although for Fe\(_5\)Sn\(_3\) there is not much difference in energy between spin directions. We note that these room temperature results (planar easy axis) for Fe\(_3\)Sn\(_2\) disagree with older data reported for polycrystalline samples\(^{10,11}\).

The magnetic anisotropy energy (MAE) is a small energy (~1 meV/atom) that often can be manipulated through alloying. As described previously, the effects of various chemical substitutions on the MAE were first calculated from first principles. The calculated results of some of the substitutions on the both the sign and the magnitude of the MAE are shown in Table 1. These calculations indicate, for example, that replacing 25% percent of the Sn with Sb should flip the easy axis from planar to uniaxial.

Based on this theoretical guidance a variety of Fe\(_3\)Sn alloys with compositions Fe\(_{3-x}\)M\(_x\)Sn\(_{2-2x}\)As\(_x\), where M = Co, Mn and A = Sb, As, Bi, Pb were prepared and evaluated. The substitution of a small amount of Sb (y < 0.1) to the Fe\(_3\)Sn structure resulted in the destabilization of the desired phase and the formation of the Fe\(_5\)Sn\(_3\) structure and a Fe\(_{0.96}\)Sn\(_{0.04}\)Sb\(_{0.05}\) alloy, with some Sb going into the Fe\(_5\)Sn\(_3\) structural phases containing iron. Only Mn could be readily substituted into the Fe\(_3\)Sn phase, as also was known from a previous report on the properties of Mn rich alloys (x > 1.5) with the Fe\(_3\)Sn\(_2\) structure\(^{12}\). All of the dopants investigated, except Mn, add electrons to the Fe\(_3\)Sn structure, which suggest that perhaps other dopants, like Sb, could be substituted for Sn if part of the Fe is replaced by Mn. Indeed, this was found to be the case and up to ~15% Sb could be substituted for Sn by co-doping with Mn (Figure 6). However, the substitution of large amounts of Mn (x = 1.5) lowered both the Curie temperature and the saturation moment (Table 3). Although the anisotropy energy of the co-doped alloys is less than pure Fe\(_3\)Sn, it still has the undesired basal plane anisotropy (Figure 7).

### Conclusion

Although Fe\(_3\)Sn has many of the desirable properties for a permanent magnet phase, the easy axis of magnetization at room temperature is in the hexagonal plane and remained so for all of the experimentally investigated chemical substitutions. Theory predicts, however, that uniaxial anisotropy can be achieved by certain substitutions specifically, alloying Sb on the Sn site. It will be of interest to study this material to determine whether a related alloy with significant (~25% or greater) Sb substitution can be found and if this is feasible to study the magnetic properties of such material.

### Methods

**Synthesis**

All of the alloys with the Fe\(_3\)Sn type structure were prepared by solid-state reaction. Powders of Sn and Fe of the appropriate weights were ground together, pressed into a pellet and sealed under vacuum in a silica ampoule. Upon the first firing, the pressed pellets were heated to 805 C at 2 C/min and held for 2 days and then quenched into water. The pellets were reground and then loaded into a furnace that was preheated to 805 C, left for another 2-3 days and then quenched into water. All alloys of the Fe\(_5\)Sn\(_3\) phase (e.g. Fe\(_{1.5}\)Mn\(_{1.5}\)Sn\(_{0.9}\)Sb\(_{0.1}\)) were prepared in the same way. Single crystals (mm size) of Fe\(_{3}\)Sn\(_2\) and Fe\(_{5}\)Sn\(_3\) could be grown out of excess Sn by co-doping with Mn (Figure 6). However, the substitution of large amounts of Mn (x = 1.5) lowered both the Curie temperature and the saturation moment (Table 3).

### Characterization

Powder x-ray diffraction measurements were performed with a PANalytical X'Pert diffractometer and a position sensitive detector using monochromated Cu K\(_\alpha\) radiation. Single crystal X-ray diffraction measurements on Fe\(_3\)Sn\(_2\) crystals were performed on a Bruker SMART APEX CCD-based single crystal diffractometer.

### Table 3: Magnetic properties of several Fe-Sn based alloys. The anisotropy energy, \( K_1 \), was estimated using the approximation \( K_1 \approx H_u M_u / 2 \). The quench temperatures for the Fe\(_5\)Sn\(_3\) compounds are shown in parenthesis.

| Composition                  | \( T_e \) (K) | \( M_1 / (\text{emu/g}) \) (300 K) | \( M_1 / (\text{emu/g}) \) (MA) | \( M_1 / (\text{emu/g}) \) (5 K) | \( M_1 / (\text{emu/g}) \) (5 K) | \( K_1 \) (test) (MJ/m\(^3\)) |
|-----------------------------|--------------|-----------------------------------|--------------------------------|-------------------------------|-------------------------------|-----------------------------|
| Fe\(_3\)Sn                  | 725          | 2.37                              | 138                            | 1.18                          | 1.23                          | 1.8                         |
| Fe\(_{4.5}\)Mn\(_{0.5}\)Sn\(_{0.95}\)Sb\(_{0.05}\) | 645          | 2.08                              | 122                            | 1.02                          | 1.11                          | 1.15                        |
| Fe\(_{1.5}\)Mn\(_{1.5}\)Sn\(_{0.9}\)Sb\(_{0.1}\)    | 405          | 1.42                              | 83                             | 0.68                          | 0.91                          | >0                         |
| Fe\(_{1.5}\)Mn\(_{1.5}\)Sn\(_{0.85}\)Sb\(_{0.15}\)  | 405          | 1.35                              | 79                             | 0.65                          | 0.90                          | >0                         |
| Fe\(_3\)Sn\(_2\) (1100 K)    | 605          | 1.84                              | 81                             | 0.70                          | 0.77                          | -                           |
| Fe\(_3\)Sn\(_2\) (1155 K)    | 635          | 1.97                              | 87                             | 0.75                          | 0.83                          | -                           |
| Fe\(_5\)Sn\(_3\) Crystal (1100 K) | 620        | 1.84                              | 81                             | 0.70                          | 0.77                          | 0.35                        |
X-ray diffractometer with Mo Kα radiation. (Additional information about the single crystal data collection and refinement process is given in the supplementary information). A Quantum Design Magnetic Property Measurement System (MPMS) was used for magnetic measurements. Because of the large magnetic signal from the ferromagnets, the typical sample mass was restricted to less than 10 mg. The sample weights were measured carefully using a balance with a sensitivity of 0.01 mg. The Curie temperatures of the ferromagnets were determined using the furnace option of the MPMS and a measuring field of 100 Oe.

Theoretical. Our calculations for Fe$_3$Sn were done using LAPW sphere radii of 2.41 Bohr for Sn and 2.35 Bohr for Fe. The internal atomic coordinates relaxed until Hellman-Feynman forces were less than 2 mRyd/Bohr. Spin-orbit coupling was included for all calculations, excepting this structure optimization. Well converged basis sets based on a cut-off $R_k$ of 9.0 were employed, with as many as 240,000 k-points in the full Brillouin zone used for the calculations of magnetocrystalline anisotropy energy (MAE). Convergence with respect to k-point sampling was carefully checked and variations in MAE at the largest mesh used were no more than a few percent. Orbital moments were also computed. These were approximately 0.073 μB per Fe and a much smaller value for Sn.

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Author contributions
B.C.S. conceived the experiments, prepared all of the samples, and made all of the magnetic measurements and analysis. M.A.M. and B.S. did all of the detailed crystallography measurements and analysis. D.J.S. and D.S.P. did all of the theoretical calculations. B.C.S. wrote the first draft of the paper, and all authors contributed to improving the manuscript.

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