Synthesis and organic field effect transistor properties of isoindigo/DPP-based polymers containing a thermolabile group†

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Introduction

For the past decade, isoindigo (ID) and diketopyrrolopyrrole (DPP) based \( \pi \) conjugated polymers (CPs) have attracted much attention as electronic materials for solution processing polymer solar cells (PSCs)\(^1\) and organic field-effect transistors (OFETs).\(^1\) As is well known, ID and DPP possess interesting properties, such as a highly fused conjugated structure, electron-deficient nature and easy synthesis. These characteristics are reflected in their polymers, which show low bandgaps, strong \( \pi-\pi \) interactions, high crystallinity and high charge carrier mobility. In addition, ID and DPP based polymers have alky1 chains on their N atoms to induce solubility.

However, CPs based on the two units exhibit different optical and electrochemical properties, such as the highest occupied molecular orbital (HOMO), the lowest unoccupied molecular orbital (LUMO) and band gaps. To date, the donor–acceptor (D–A) system, the alternative linkage between electron-rich and electron-deficient aromatic units, is one of the important protocols to design low bandgap electron donating materials in organic photovoltaic cells (OPVs),\(^2\) as well as high mobility CPs for OFETs.\(^3\) Careful design and the matching between the selected donor and acceptor molecular units help us to tune the HOMO and LUMO energy levels as well as the bandgap of synthesized CPs, owing to the HOMO and LUMO energy levels as well as the bandgap of the donor and acceptor moieties, respectively.\(^4\) For example, isoindigo-based low bandgap polymers, containing thiophene as a donor moiety (PT-ID1), were synthesized and the isoindigo monomer was branched with a 2-octyldecyl side chain to improve the solubility.\(^5\) The HOMO and LUMO energy levels of PT-ID1 were –5.49 and –3.91 eV, respectively. A photovoltaic device with PT-ID1/PC\(_{71}\)BM in a 1:2 ratio as the active layer (70 nm) gave an open circuit voltage (\( V_{oc} \)) value of 0.87 V, a fill factor (FF) of 60%, and a short circuit current (\( j_{sc} \)) value of 1.76 mA cm\(^{-2}\), resulting in a power conversion efficiency (PCE) value of 0.92%. The same polymer structure on the application of OFETs showed a hole-mobility...
characteristics, which may facilitate the hole transport a range from 1.38 to 1.46 eV and have semi-crystalline structure.

A low HOMO level, which is essential for high performance, was recovered.

(boc) protecting group is a convenient group for protecting molecule materials, such as CPs and small molecules based on Boc-

protected electron dection and has been employed to allow solution processing.

All the previous CPs contain alkyl side chains for solubility. Recently, we have synthesized several novel organic materials recently, we have synthesized several novel organic materials under highly dry conditions, a mixture of 5-bis(triethylstannyl) thiophene (5) (0.66 g, 1.00 mmol), the desired alkylated dibromo-derivative of ID/DPP materials and the desired Boc-derivative of ID/DPP materials (2 and/or 4) (0.2 mmol) and Pd(PPh3)4 (0.058 g, 0.05 mmol) in dry DMF (20 mL) was degassed with nitrogen for 20 min then heated under reflux for 24–48 h at a temperature of 120 °C. The reaction mixture was allowed to return to room temperature and then poured into methanol. The crude polymer was collected by filtration and washed repeatedly with methanol. The residual solid was loaded into an extraction thimble and washed with methanol (24 h) followed by acetone (24 h).

**Experimental**

**Synthetic details**

Co-monomers, \((E)-6,6^′\)-dibromo-1,1-bis(2-octyldodecyl)(3,3′-biindolylindene-1)-2,2′-dione (1) 2,5-bis(2-octyldodecyl)-3,6-di(5-bromothien-2-yl)pyrrolo[3,4-c]pyrrole-1,4-(2H,5H)-dione (3) and 2,5-bis(tert-butyloxycarbonyl)-3,6-di(5-bromothien-2-yl)pyrrolo[3,4-c] pyrrole-1,4-(2H,5H)-dione (4) were synthesized as per the published literature.1–3

1,1-Bis(tert-butyloxycarbonyl)-(3,3′-biindolylindene)-2,2′-dione (Boc ID) (2). A solution of \((E)-6,6^′\)-dibromo-1H,1′H[3,3′] biindolyliden-2,2′-dione (1.00 g, 2.40 mmol) and a catalytic amount of 4-(dimethylamino)pyridine (DMAP) in dry THF (100 mL) was stirred at room temperature under argon atmosphere. After 20 min, di-tert-butyl dicarbonate (Boc2O) (1.60 g, 7.10 mmol) was added dropwise over 30 min. The solution was stirred at 35 °C overnight, and then the solvent and excess of Boc2O were removed under reduced pressure at 45 °C to give 2 (1.34 g, yield: 90%) as a bright red solid.1 H NMR (CDCl3, 400 MHz), δ (ppm): 8.67 (d, J = 8.4 Hz, 2H), 7.98 (d, J = 7.16 Hz, 2H), 7.22 (d, J = 8.5 Hz, 2H), 1.92 Hz, 2H), 6.81 (d, 2H), 1.60 (s, 18H, Boc-H).13 C NMR (CDCl3, 100 MHz), δ (ppm): 165.33, 148.33, 141.88, 131.37, 130.14, 127.26, 125.49, 120.42, 117.85, 85.50, 30.28, 28.06. Elemental analysis: found: C, 50.12; H, 3.91; Br, 25.63; N, 4.50; O, 15.32%. Calc. for C26H24Br2N2O6: C, 50.34; H, 3.90; Br, 25.76; N, 4.52; O, 15.48%.

**General procedures for Pd catalyzed Stille coupling arylation for the synthesis of the new polymers**

Under highly dry conditions, a mixture of 5-bis(triethylstannyl) thiophene (5) (0.66 g, 1.00 mmol), the desired alkylated dibromo-derivative of ID/DPP materials (1 and/or 3) (0.8 mmol), the desired Boc-derivative of ID/DPP materials (2 and/or 4) (0.2 mmol) and Pd(PPh3)4 (0.058 g, 0.05 mmol) in dry DMF (20 mL) was degassed with nitrogen for 20 min then heated under reflux for 24–48 h at a temperature of 120 °C. The reaction mixture was allowed to return to room temperature and then poured into methanol. The crude polymer was collected by filtration and washed repeatedly with methanol. The residual solid was loaded into an extraction thimble and washed with methanol (24 h) followed by acetone (24 h).

**P[ODIDT-BID]** dark violet color. 1 H NMR (CDCl3, 400 MHz), δ (ppm): 9.27–8.92 (br, 4H), 7.49–6.48 (br, 10H), 3.81–3.33 (br, 2H), 2.72–2.49 (br, 4H) 2.06–1.78 (br, 22H), 1.58–1.05 (br, 76H), 1.02–0.72 (br, 18H).

**P[ODIDT-BDPP]** dark violet color. 1 H NMR (CDCl3, 400 MHz), δ (ppm): 9.31–8.94 (br, 4H), 7.47–6.51 (br, 10H), 3.89–3.48 (br, 2H), 2.70–2.58 (br, 4H) 2.07–1.81 (br, 22H), 1.50–1.04 (br, 76H), 0.96–0.71 (br, 18H).

**P[ODDPPt-BID]** dark violet color. 1 H NMR (CDCl3, 400 MHz), δ (ppm): 9.00–8.92 (br, 4H), 7.55–6.79 (br, 10H), 4.14–3.99 (br, 2H), 2.00–1.95 (br, 4H), 1.82–1.59 (br, 22H), 1.59–1.19 (br, 76H), 0.98–0.71 (br, 18H).

**P[ODDPPt-BDPP]** dark violet color. 1 H NMR (CDCl3, 400 MHz), δ (ppm): 9.09–8.76 (br, 4H), 7.40–6.73 (br, 10H), 4.06–3.37 (br, 2H), 2.08–1.84 (br, 4H), 1.73–0.95 (br, 98H), 0.93–0.71 (br, 18H).
Device fabrication

The bottom-gate/bottom-contact (BGBC) OFET devices were fabricated using n++-Si/SiO2 (200 nm) substrates. The Au electrodes (40 nm) for the source and drain were thermally evaporated with a length of 50 μm and a width of 1000 μm on the SiO2 surface under high vacuum (~10⁻⁶ Torr). The substrates were subjected to cleaning by ultra-sonication in deionized water, acetone, and isopropanol each for 20 min. The cleaned substrates were dried in an oven at 80 °C for 2 h. The substrates were UV/ozone treated for 20 min and then transferred into a nitrogen-filled glovebox. The thin films of the random CPs were spin-cast on the substrates with the random CP solutions (1 wt% in chloroform) at 3000 rpm for 30 s. As-prepared films were annealed at 200 °C for 10 min.

Results and discussion

Molecular design and synthesis

Random donor-acceptor polymers based on diketopyrrolopyrrole/isodindigo-thiophene were selected to be the semiconducting backbones for several reasons, such as the excellent device performance that has been shown in organic electronic devices and the easy functionalization of the nitrogen atom on the isodindigo or diketopyrrolopyrrole units. To ensure solubility a large number of t-Boc and octyldodecyl groups were attached. In previous work, the thermolabile random copolymerization strategy showed good solubility with high stacking order in the annealed solid state but had a possible low impact on the electronic performance. 28

(E)-6,6′-Dibromo-1,1-bis(2-octyldodecyl)-(3,3′-biindolinyldiene)-2,2′-dione (1) was reported in our previous work (Fig. 1). 28 In this work, we introduce (E)-6,6′-dibromo-1,1-bis[tert-butyloxy- carbonyl]-(3,3′-biindolinyldiene)-2,2′-dione (Boc ID) (2) by the modification of the reported procedure: 27 treatment of (E)-6,6′- dibromo-1H,1H-[3,3′]biindolylidene-2,2′-dione with 3 equivalents of tert-butyl dicarbonate (BocO) in tetrahydrofuran (THF) at room temperature resulted in over 80% pure product. Two precursor copolymers, 2,5-bis(2-octyldodecyl)-3,6-di[5-bromothien-2-yl]pyrrole[3,4-c]pyrrole-1,4-(2H,5H)-dione (3) and 2,5bis(tert-butyloxy carbonyl)-3,6-di[5-bromothien-2-yl]pyrrole[3,4-c]pyrrole-1,4-(2H,5H)-dione (4), were synthesized as per the published literature, 27,28 (Fig. 1). Moreover, lithiation of thiophene by n-BuLi at −80 °C followed by quenching with tributyltinchloride gave the corresponding 2,5-bis(tributylstannyl) thiophene (5). All materials were extracted with organic solvents and highly purified via column chromatography.

Synthesis of random copolymers

Based on Stille coupling conditions, the polymerization reactions were done in degassed anhydrous N,N-dimethylformamide (DMF) as a solvent in the presence of catalytic amounts of Pd(PPh3)4 under N2 atmosphere at 80 °C using thermal methodology. The alkylated acceptors and t-butoxy carbonyl moieties were mixed with 2,5-bis(tributylstannyl) thiophene (5) in a molar ratio of 0.8 : 0.2 : 1, respectively.

New random conjugated polymers were designed to obtain large numbers of octyldodecyl comonomers to insure the solubility for film casting. Polymers P(ODIDT-BID) and P(ODIDT-BDPP) were obtained via Stille coupling reaction, in which comonomers 1 and distannyl derivative 5 were mixed with Boc monomers 2 or 4, respectively, in good yields (77% and 73% respectively) and acceptable average molecular weights (Mn) and reasonable polydispersity index (PDI) values (Table 1). Similarly, comonomers 3 and distannyl derivative 5 were reacted with Boc monomers 2 or 4 under Stille coupling conditions in DMF to afford P(ODDPPT-BID) and P(ODPPPT-BDPP) random copolymers in 75% and 73% yields, respectively, and showed reasonable molecular weights (Table 1). The synthetic route to the polymers is shown in Scheme 1.

All copolymers showed good solubility at room temperature in most common organic solvents. After purification by Soxhlet extraction with methanol and acetone repeatedly to remove the by-products and the oligomers, all polymer structures were confirmed by 1H NMR and elemental analyses. All the data were inserted with the proposed structures (see the ESI† for the 1H NMR spectra).

Thermal properties

Under nitrogen atmosphere, thermogravimetric analysis (TGA) was used to investigate the thermal properties of the four new CPs. From TAG analysis, all polymers retained 50% residual weights when the temperature increased up to 800 °C, except P(ODIDT-BDPP), which lost 60% of its total weight at 750 °C. All polymers showed two decomposition steps. The thermal decomposition related to isolation of the tBoc function appeared at Td 98.8–99.8 wt% residues in the range of 180–220 °C.

The second thermal decomposition temperatures (Te, 92–84 wt% residues) were in the range of 388–436 °C, related to the decomposition of the long alkyl chains during heating, 29 which confirmed the high thermal stability of all polymer products. The thermogravimetric analysis diagrams of the synthesized random copolymers are shown in the ESI (Fig. S17). To provide the formation of H-bonds after the thermal elimination of the t-Boc groups, Fourier transform infrared (FT-IR) spectroscopy was used (Fig. S2†). In general in the annealed films, a weak band was found at around 3450 cm⁻¹, which was assigned to the lactam structure N–H…O=C hydrogen bonding of the isodindigo and DPP units. In addition, the stretching vibration band of C==O at ~1700 cm⁻¹ disappeared, which clearly confirmed the complete decomposition of t-Boc groups, and the lactam moiety C==O shifted to slightly smaller wavenumber.
Optical properties

Fig. 2 shows the solution and film absorption spectra of the new polymers. In dilute solutions of polymers in chloroform, \( P(ODIDT \cdot BID) \) and \( P(ODDPPT \cdot BDPP) \) offered one absorption band, which indicates the intermolecular charge transfer (ICT) between donor and acceptor units, with maxima at 612 and 616 nm, respectively.

| Polymer                  | \( M_n (\text{kg mol}^{-1}) \) | PDI \( (M_w/M_n) \) | Yield\(^c\) (%) | \( T_d \)\(^d\) (°C) |
|--------------------------|-------------------------------|---------------------|----------------|------------------|
| \( P(ODIDT \cdot BID) \) | 20.2                          | 1.81                | 77.0           | 338.4            |
| \( P(ODIDT \cdot BDPP) \) | 19.7                          | 1.77                | 73.0           | 390.3            |
| \( P(ODDPPT \cdot BID) \) | 16.5                          | 1.69                | 75.0           | 436.2            |
| \( P(ODDPPT \cdot BDPP) \) | 17.2                          | 1.71                | 73.0           | 391.9            |

\(^a\) All polymerizations were carried out using the Stille cross-coupling method. \(^b\) Calculated from GPC (eluent, CHCl\(_3\); 30 °C; polystyrene standards). \(^c\) Based on the weight of pure polymer obtained. \(^d\) Determined by DSC under nitrogen atmosphere at a heating rate of 10 °C min\(^{-1}\).
However, P(ODIDT-BID) and P(ODIDT-BDPP) showed two major absorption bands in dilute solutions, the shorter wavelength absorption bands at 451 and 454 nm for P(ODIDT-BID) and P(ODIDT-BDPP), respectively, are indicated from the π–π* transition and the longer absorption bands at wavelength 646 nm and 620 nm, respectively, representing the ICT at the donor–acceptor composition.

In solid films, polymers containing ODDPP showed broader absorbance than ODID-based polymers, which reflects the high π-stacking ordering in the case of ODDPP based polymers. Corresponding to the absorption edges, polymers based on ODID represent a slight red shift in solid state film compared to in dilute solutions. P(ODIDT-BID) showed λ_onset = 823 nm in film and λ_onset = 796 nm in solution. Similarly, P(ODIDT-BDPP) exhibits absorption edges at 850 nm in film and 792 nm in solution. In contrast, random CPs based on ODDPP offer long range red shift and broadness compared to their corresponding absorptions in solution, reflecting the increasing interchain interaction and ordering along the polymer backbones as well as the extended π-conjugation length in the solid state. The onset absorptions of P(ODDPPT-BID) and P(ODDPPT-BDPP) are 1036 nm and 980 nm, respectively, more red shifted by approximately 200 nm than the absorptions edges in solution. The optical bandgaps of all polymers, which were calculated from λ_onset in the solid state \[ E_{\text{op}}^\text{g} (eV) = \frac{1240}{\lambda_{\text{onset}} (nm)} \], are 1.51, 1.46, 1.20 and 1.26 eV for P(ODIDT-BID), P(ODIDT-BDPP), P(ODDPPT-BID) and P(ODDPPT-BDPP), respectively (Table 2). The new random CPs possess lower optical band gaps and broader absorption windows than their corresponding CPs not containing N-block tBoc-acceptors along their backbone structures. Interestingly, in the as-cast films, P(ODDPPT-BID) and P(ODDPPT-BDPP) offered second short absorption bands, which indicate the π–π* transition, at 423 nm and 420 nm, respectively.

Furthermore, after annealing polymer films up to 220 °C to ensure the recovery of the N–H function and liberation of the tBoc group, all polymers showed two major absorption bands and induced a slight red shift, which indicates a great increase of long range ordering and the π-stacking orientation of the semi-crystalline polymer structures.

**Electrochemical properties**

The electrochemical properties of the polymers were explored by cyclic voltammetry (CV). The cyclic voltammograms of the

**Table 2** Optical properties of the four copolymers

| Polymer          | λ_max (nm) | λ_max (nm) | λ_onset (nm) | E_{\text{op}}^\text{g} (eV) |
|------------------|------------|------------|--------------|-----------------------------|
| P(ODIDT-BID)     | 641        | 641        | 791          | 1.57                        |
| P(ODIDT-NHID)    | 641        | 641        | 801          | 1.55                        |
| P(ODIDT-BDPP)    | 617        | 680        | 801          | 1.55                        |
| P(ODIDT-NHDP)    | 684        | 684        | 807          | 1.54                        |
| P(ODDPPT-BID)    | 614        | 839        | 1019         | 1.22                        |
| P(ODDPPT-NHID)   | 782        | 782        | 1038         | 1.20                        |
| P(ODDPPT-BDPP)   | 618        | 751        | 955          | 1.30                        |
| P(ODDPPT-NHDP)   | 744        | 744        | 955          | 1.30                        |
P(ODDT-BID), P(ODDT-BDPP), P(ODDPPT-BID) and P(ODDPPT-BDPP) films are shown in Fig. 3.

The HOMO and LUMO energy levels of the polymers were calculated according to the equations: $E_{\text{HOMO}} = -(E_{\text{ox}} + 4.71) \text{ eV}$ and $E_{\text{LUMO}} = -(E_{\text{red}} + 4.71) \text{ eV}$, where $E_{\text{ox}}$ and $E_{\text{red}}$ are the onset oxidation and reduction potentials, respectively, of the polymers vs. Ag/AgCl (saturated KCl) as a reference electrode. The HOMO energy levels moved gradually deeper in the order $[P(ODDT-BID) > P(ODDT-BDPP) > P(ODDPPT-BID) > P(ODDPPT-BDPP)]$ and showed $-5.44, -5.65, -5.68$ and $-5.69 \text{ eV}$, respectively. However, there are contrasts in the LUMO energy levels of the polymers. The random CPs based on the same structure of electron accepting units offered higher LUMO energy levels than the other containing a mixed (1D/DPP) backbone structure. The LUMO energy level of P(ODDPPT-BDPP) was the highest at $-3.89 \text{ eV}$, thus P(ODDT-BID) and P(ODDPPT-BDPP) exhibited LUMO energy levels at $-3.97$ and $-4.13 \text{ eV}$, respectively. However, the LUMO energy level of P(ODDPPT-BDPP) was the lowest at $-4.17 \text{ eV}$. Moreover, the electrochemical bandgaps ($E_g = E_{\text{HOMO}} - E_{\text{LUMO}}$) of P(ODDT-BID), P(ODDT-BDPP), P(ODDPPT-BID) and P(ODDPPT-BDPP) were 1.47, 1.52, 1.51 and 1.80 eV, respectively.

In the case of decarboxylated polymers (NH-forms), the improvement of $\pi-\pi$ packing did not show a significant change to the HOMO and LUMO energy levels. However, in the case of P(ODDT-NHID), it showed a deeper HOMO level at $-5.62 \text{ eV}$ than its corresponding P(ODDT-BID) polymer.

In many aspects, the new synthesized random CPs showed low band-gaps closely matched the ideal low band gaps of the polymers used for organic electronic applications.

**Crystalline structure of polymers**

Charge carrier transport strongly depends on microstructure in CPs. To explain that, the X-ray diffractometry (XRD) measurements of the as-spun films and films annealed at 200 °C are shown in Fig. 4. The as-spun films of P(ODDPPT-BID) and P(ODDPPT-BDPP) showed diffraction peaks at $2\theta = 6.65^\circ$ and $6.51^\circ$, which corresponded to $d$-spacings of 13.3 Å and 13.6 Å, respectively. However, P(ODDT-BID) and P(ODDT-BDPP) exhibited weak arrangement of diacid units, for $2\theta = 4.00^\circ$ and $3.87^\circ$, proportional to $d$-spacings of 22.1 Å and 22.8 Å, respectively, considered to reflect the inter-lamellar distance induced by octyldodecyl chains (Fig. 4a). After annealing at 200 °C for 10 min to remove the Boc function, the molecular ordering was improved by thermal annealing as well as the formation of H bonding across polymer chains. As a result, two new sharp and intensive diffraction peaks were found after the annealing process. The diffraction peaks at $2\theta = 8.52^\circ$, $8.56^\circ$, $8.74^\circ$ and $8.67^\circ$, for P(ODDT-BID), P(ODDT-BDPP), P(ODDPPT-BID) and P(ODDPPT-BDPP) films, respectively, considering the $d$-spacings of 10.0–10.4 Å confirming much stacking between the alkyl chains, owing to the formation of H bonding between polymer chain upon the removal of the Boc function during heating, which resulted in the shortening of the interlayer distance.

The other diffraction peaks at $2\theta = 17.28^\circ$, $17.32^\circ$, $17.98^\circ$ and $17.39^\circ$ for P(ODDT-BID), P(ODDT-BDPP), P(ODDPPT-BID) and P(ODDPPT-BDPP) films, respectively, corresponding to $d$-spacings of 5.0–5.2 Å, indicating that it adopted the highest degree of $\pi-\pi$ stacking, suggested that these peaks did not only originate from crystallites formed by a combination of edge-to-edge and face-to-face packing but also from localized aggregation. These localized aggregates could readily form in synthesized polymers because of the enhanced backbone planarity and the incorporated successive H-bonding, which exhibit greatly reduced steric hindrance compared with Boc units.

**Field-effect properties of polymers**

To investigate the charge transport properties of the synthesized random CPs, we fabricated bottom-gate and top-contact (BGTC) OFETs with them. Fig. 5b presents the transfer curves of the OFETs with the random CPs. There are no field-effect characteristics in the devices with P(ODDT-BID) and P(ODDT-BDPP), while the P(ODDPPT-BID) and P(ODDPPT-BDPP) show fine field-effect characteristics. The field-effect mobilities were extracted from the saturation region in the transfer curves by using the following equation: $I_{DS} = \frac{W}{2L}C_{GL}(V_{GS} - V_T)^2$ (Table 3), where $W$, $L$, $C$, $\mu$, $I_{DS}$, $V_{GS}$ and $V_T$ are channel width,
channel length, gate dielectric capacitance per unit area, charge carrier mobility, drain-source current, gate-source voltage, and the threshold voltage, respectively.

As summarized in Table 3, as-spun films of P(ODIDT-BID) and P(ODIDT-BDPP) exhibited relatively lower hole mobilities of \(2.26 \times 10^{-8} \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}\) and \(6.68 \times 10^{-8} \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}\), respectively, while higher hole mobilities were obtained in the as-spun films of P(ODIDT-BID) and P(ODIDT-BDPP) (1.17 \(\times 10^{-3}\) cm² V⁻¹ s⁻¹ and 1.41 \(\times 10^{-3}\) cm² V⁻¹ s⁻¹, respectively); these results are in good agreement with the observations from the XRD analysis. However, after the films were annealed at 200 °C, poor performance with relatively lower \(\mu\) and negatively-shifted \(V_T\) were observed, which can be ascribed to the interruption of charge carrier injection owing to the morphology change during thermal annealing.

Surface morphologies
Therefore, we examined the surface morphologies of the films before and after annealing with tapping mode atomic force microscopy (AFM).

![Fig. 4 XRD diagrams obtained from all polymer thin films on ITO substrate (a) as-spun thin film (r.t.) (b) annealed at 220 °C for 30 min, and (c) schematic representations of \(\pi-\pi\) stacking P(ODIDT-BID) as cast and after annealing.](image)

![Fig. 5 Transistor characteristics of synthesized new polymers: (a) illustration of OFET configuration. (b) and (c) Transfer curves of devices with P(ODIDT-BID), P(ODIDT-BDPP), P(ODDPP-T-BID) and P(ODDPP-T-BDPP) (b) as casted and (c) after annealing treatment at 200 °C.](image)
As shown in Fig. 6, the pristine films of P(ODIDT-BID) and P(ODIDT-BDPP) showed smooth surface morphology, but that did not reveal any structural features. However, densely-packed large domains but rough morphologies were observed in P(ODDPPT-BID) and P(ODDPPT-BDPP) films owing to relatively strong inter-lamellar stacking. After annealing, no significant change was observed in the morphology of the films of P(ODIDT-NHID) and P(ODIDT-NHDPP), compared to those of P(ODIDT-BID) and P(ODIDT-BDPP) films. Note that P(ODDPPT-NHID) and P(ODDPPT-NHDPP) films exhibited smaller aggregates or domains in the surface morphology. These phenomenon agreed with the XRD results, whereby aggregated crystallities were formed by H bonding across polymer chains. Removing tBoc-units by thermal annealing results in molecular ordering within small aggregates having disconnected morphology. This kind of molecular ordering leads to difficulties in charge carrier mobility in OFET devices, which require lateral transfers, but this may have better performance in vertical transfer or in other future devices.

Conclusions

Narrow bandgap random co-polymeric semiconductors, comprising alkyl substituted diketopyrrolopyrrole (DPP) and isoindigo (ID) and tert-butoxycarbonyl (t-BOC)-protected DPP/ID, were synthesized and used in organic field-effect transistors (OFETs). The polymer films were prepared by solution deposition and thermal annealing of precursors featuring thermally labile t-BOC groups. The effects of the thermal cleavage on the molecular packing structure in the polymer thin films were investigated using thermogravimetric analysis (TGA), UV-vis spectroscopy, and X-ray diffraction (XRD) analysis. By utilizing t-BOC thermolysis at 220 °C, the BOC films were directly converted to deprotected films that can potentially possess a hydrogen-bonded network. Finally, we fabricated OFET devices with these random CPs. The trend of field-effect mobility of the as-spun OFET corresponds to that of the XRD data. However, excessive aggregation of the random CP films during thermal annealing results in the performance degradation in the OFET devices.

Acknowledgements

The research was supported by the Basic Science Research Program though the National Research Foundation (NRF) funded by the Ministry of Science, ICT and Future Planning (NRF-2015R1A2A101002493); and by the GIST Research Institute (GRI) project through a grant provided by GIST in 2017.

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