Synthesis and magnetorheological effect of Fe3O4-TiO2 nanocomposite

Jianhong Wei  
*University of Wollongong*

C J Leng  
*Wuhan University, Wuhan, China*

Xianzhou Zhang  
*University of Wollongong, xianzhou@uow.edu.au*

Weihua Li  
*University of Wollongong, weihuali@uow.edu.au*

Z Y Liu  
*Wuhan University, Wuhan, China*

*See next page for additional authors*

Follow this and additional works at: [https://ro.uow.edu.au/engpapers](https://ro.uow.edu.au/engpapers)

Part of the Engineering Commons  
[https://ro.uow.edu.au/engpapers/4110](https://ro.uow.edu.au/engpapers/4110)

**Recommended Citation**  
Wei, Jianhong; Leng, C J; Zhang, Xianzhou; Li, Weihua; Liu, Z Y; and Shi, J: Synthesis and magnetorheological effect of Fe3O4-TiO2 nanocomposite 2009, 012083.  
[https://ro.uow.edu.au/engpapers/4110](https://ro.uow.edu.au/engpapers/4110)

Research Online is the open access institutional repository for the University of Wollongong. For further information contact the UOW Library: research-pubs@uow.edu.au
Synthesis and magnetorheological effect of Fe3O4-TiO2 nanocomposite

J H Wei1,2, C J Leng1, X Z Zhang2, W H Li2, Z Y Liu1 and J Shi1

1 Key Laboratory of Acoustic and Photonic Materials and Devices of Ministry of Education and Department of Physics, Wuhan University, Wuhan, 430072, P.R. China

2 School of Mechanical, Materials & Mechatronic Engineering, University of Wollongong, NSW, 2522, Australia

E-mail: jhwei@whu.edu.cn, weihuali@uow.edu.au

Abstract. Aimed to obtain a material with improved rheological property, the Fe3O4-TiO2 nanocomposites were prepared by a modified sol-gel processing. The structure and morphology of the Fe3O4-TiO2 nanocomposites were examined by transmission electron microscopy (TEM), X-ray diffraction (XRD) analysis and Fourier transform infrared spectroscopy (FT-IR) analysis, etc. The magnetic property and the magnetorheological properties of the coated particles were also examined in detail. The experimental results demonstrated that such materials exhibited favorable magnetorheological (MR) effect, and the MR performance of them were dependent on the relative content of TiO2 in the composite.

1. Introduction

The rheological properties of some fluids can be reversibly modified in the presence of an external field, such as electrorheological (ER) fluids and magnetorheological (MR) fluids. The ER fluids and MR fluids, known as smart liquids for their apparent viscosity capable of experiencing a rapid, reversible change upon application of an external electric field or magnetic field [1-3], have potentially important application in numerous electromechanical devices, such as valves, dampers, and clutches in the automotive and robotics industries, etc. Recently, Wen and other researches [4-6] reported a theoretical research on the electro-magneto-rheological (EMR) effect, the combination of an ER effect and a MR effect, which provides us with a new strategy to control the rheological properties of fluid materials. Although the results for the theoretical research were inspiring, the experimental research on EMR fluids was less reported until now.

In general, the desired particles used in MR or EMR suspensions should have both tiny or no coercive force and large saturated magnetization strength (M_s), such as carbonyl iron, ferrite (including Fe3O4) and their composites etc. [7-9] Due to large saturated magnetization strength, Fe3O4 nanoparticles have been widely used to form ferrofluids and MR fluids. However, unmodified Fe3O4 nanoparticles usually incline to aggregate into the clusters, have poor ability of anti-oxidation and poor thermal stability. To overcome this problem, the coating processing with insulating oxide on the magnetic nanoparticles was usually employed. In this paper, we report a novel composite, a simple chemical synthesis to coat Fe3O4 nanoparticles with TiO2 layer. The TiO2 dielectric coating was fulfilled by a modified sol-gel process and the thickness of coating layer was governed by solution concentrations and processing durations. The TiO2 coating on magnetic nanoparticles was beneficial to
2.1 Preparation of Fe$_3$O$_4$-TiO$_2$ nanocomposites

The nanosized Fe$_3$O$_4$ particles were prepared by a chemical coprecipitation method [10]. The Fe$_3$O$_4$ nanoparticles were coated with TiO$_2$ by a modified sol-gel method. In a typical coating procedure, a suitable amount of magnetite particles dispersed in ethanol was ultrasonic for 1h, then, Ti(OR)$_4$, polyethylene glycol, diethylyamine and water were slowly added into the mixture. The reaction was allowed at room temperature for 3h under continuous stirring. And then the reaction was continued at 85°C in a rotary evaporation meter to make the solvent fully evaporating. Finally, the resultant powder was heat treated at 100°C for 8h to remove any trace of water. Here, diethanolamine was used as an additive to prevent the precipitation of titanium butoxide from the alcoholic in the presence of excess water. PEG was added to increase the content of surface hydroxyl in the composite.

2.2 Characterization and measurement

The morphologies of the samples were investigated by transmission electron microscopy (TEM, JEM-JEOL-2010). The crystal phase of the prepared products were analyzed by X-ray powder diffraction (XRD, Model Japan Rigaku D/max-\(\gamma\)A \(K_{\alpha}\), Cu-target, \(\lambda = 0.1541\)nm). The TEM photographs and XRD patterns reveal that the Fe$_3$O$_4$ nanoparticles with diameter about 10nm were completely covered by the amorphous TiO$_2$ layers. The densities of the particles were measured with a pycnometer by dispersing particles in silicone oil. The densities of the coated particles change from 2.36 to 2.83g/cm$^3$ with different TiO$_2$ content, which are far less than the density of Fe$_3$O$_4$ 4.25g/cm$^3$. Low particle density is beneficial to the stability of MR fluids.

The chemical structure was determined by a Nicolet Dx-10 Fourier transform infrared spectroscopy (FT-IR) spectrophotometer. The hysteresis cycles of magnetite nanoparticles and Fe$_3$O$_4$-TiO$_2$ composites were recorded at room temperature by using a vibrating-sample magnetometer (VSM, EG&G Princeton Applied Research Vibrating Sample Magnetometer, Model 155). An Anton Paar Physica MCR301 rheometer in parallel-plate configuration was used to measure rheological properties of the composite. All the experiments were conducted at the room temperature.

3. Results and Discussion

3.1 Structure and magnetic properties

Figure 1 shows the FT-IR spectra of (a) the mixture of Fe$_3$O$_4$ and TiO$_2$ (b) and Fe$_3$O$_4$-TiO$_2$ nanocomposites respectively. The spectrum (a) shows that the broad band around 3413cm$^{-1}$ is the asymmetric and symmetric stretching vibrations of O-H group, whereas the band around 1636cm$^{-1}$ is the H-O-H bending of the coordinated water, the band at 1451 cm$^{-1}$ and 1384 cm$^{-1}$ is the bending vibrations of C-H group, the band at 1087 cm$^{-1}$ and 950cm$^{-1}$ is the characteristic vibrations of Ti-O-C group, and the band at 500-700 cm$^{-1}$ is attributed to the Ti-O-Ti or Fe-O stretching vibrations. In comparing the two spectra in Figure 1, it is noted that although the two spectra are similar as a whole, there are some observable differences as marked on spectrum(b), the band at 3413cm$^{-1}$ in spectrum(a) shift to 3385 cm$^{-1}$ in spectrum (b) , and getting into broader, whereas the band at 1636 cm$^{-1}$ in spectrum (a) shift to 1627 cm$^{-1}$ in spectrum (b) and a new peak appeared at around 890 cm$^{-1}$. So, we
infer there must exist some interactions between the Fe-OH and Ti-O-Ti band, the interaction make them uneasy to separate.

Figure 1. FT-IR spectra for (a) the mixture of Fe$_3$O$_4$ and TiO$_2$ (b) Fe$_3$O$_4$-TiO$_2$ nanocomposites

The hysteresis loops of Fe$_3$O$_4$ nanoparticles and Fe$_3$O$_4$-TiO$_2$ nanocomposites are shown in Figure 2. For Fe$_3$O$_4$ nanoparticles, saturated magnetization (M$_s$), remnant magnetization (M$_r$) and coercive force(H$_c$) are estimated to be M$_s$= 41.6emu/g, M$_r$=0, and H$_c$=0. No hysteresis loop is observed, indicating a super paramagnetic behavior. For composite structures with 1ml Ti(OR)$_4$ precursor, its saturated magnetization was 12.9emu/g. With the increasing of the TiO$_2$ content in the composite, the saturation magnetization of the particle decreases. Moreover, these saturation magnetization values are inevitably much lower than that of pure Fe$_3$O$_4$ particles, which is consistent with the mass proportion of the magnetite nanoparticles in the coated particles. In addition, Fe$_3$O$_4$-TiO$_2$ nanocomposites also show super paramagnetic character. These particles having no coercivity seem to give a distinct switching response of the MR effect if the field is switched on and off, respectively, which is beneficial to its applications in MR fluids.

3.2 MR effect

MR fluids of pure Fe$_3$O$_4$ and Fe$_3$O$_4$-TiO$_2$ nanocomposites at a concentration of 30 wt% in silicone oil were prepared. Rheological behavior of MR suspensions was investigated under steady shear flow and static magnetic field. Figure 3 shows the response of shear stress with shear rate for the Fe$_3$O$_4$-TiO$_2$ nanocomposites based suspension with and without magnetic field. The induced magnetic moment of the particle caused by an applied magnetic field is parallel to the field direction. The MR suspension exhibited significant rheological properties in the range of shear rate from $1/1$s to $100/1$s. The flow behavior of suspensions made of Fe$_3$O$_4$-TiO$_2$ nanocomposites only shows a slight departure from Newtonian fluid without the magnetic field. With the application of magnetic field, the MR fluids exhibit the Bingham plastic behavior and the shear stress increases with the increasing of the magnetic field strength.

Figure 4 shows the effects of the TiO$_2$ contents on MR properties of the Fe$_3$O$_4$-TiO$_2$ nanocomposites based MR fluids. It can be seen that the MR fluid of the composites with 1ml Ti(OR)$_4$ precursor showed the largest shear yield stresses over the other composites based MR fluid due to the higher iron oxide content. However, the composites based MR fluids show significant improved shear stress compared with that of pure magnetite nanoparticles. The possible reason maybe the outer
TiO$_2$ layer inhibiting the direct contact of Fe$_3$O$_4$ particles, which facilitate the particle orientation and the rearrangement upon a magnetic field and make it easy to recover the normal state of dispersion after the removal of the field. In addition, sufficient active groups on the surface of particles promote the surface activity, which further promote the MR performance. However, with the increasing of TiO$_2$ content in coated particles, the magnetic property decreases, which reduced the MR performance accordingly. So, in this report, the MR fluid of the composites with 1ml Ti(OR)$_4$ precursor showed the optimum shear yield stresses over the other Fe$_3$O$_4$-TiO$_2$ nanocomposites based MR fluid.

**Figure 3.** Influence of shear rate on shear stress for Fe$_3$O$_4$-TiO$_2$ nanocomposites based MR fluids under various magnetic fields

**Figure 4.** Effects of the TiO$_2$ contents on MR properties of the Fe$_3$O$_4$-TiO$_2$ nanocomposites based MR fluids.

4. Conclusions
A novel composite, Fe$_3$O$_4$-TiO$_2$ (TiO$_2$-coated Fe$_3$O$_4$) nanocomposites were prepared by modified sol-gel method. The FT-IR spectra infer the existence of interaction between Fe$_3$O$_4$ and TiO$_2$. The Fe$_3$O$_4$-TiO$_2$ nanocomposites show a superparamagnetic, without coercivity or remanence at room temperature, the magnetic property of the particles can be adjusted by the relative content of Fe$_3$O$_4$ and TiO$_2$. The MR fluids containing such materials in silicone oil show significant improved shear stress compared with that of pure magnetite nanoparticles, the shear stress of composite based MR fluids decreases with the increasing of the relative content of TiO$_2$.

Acknowledgments: We are grateful for the financial support from the National Program on key Basic Research Project (973 Grant No. 2009CB939705) and the National Natural Science Foundation of China (Grant No. 10674105 and 10874131).

References:
[1] Parthasarathy M and Klingenberg D J 1996 Mater. Sci. Eng. R: Reports, 17 57
[2] Ma H R, Wen W J, Tam W Y and Sheng P 2003 Adv. Phys. 52 343
[3] Tao R 2001 J. Phys. Condens. Mater. 13 R979
[4] Wen W J, Wang N, Ma H R, Lin Z F, Chan C T and Sheng P 1999 Phys. Rev. Lett. 82 4248
[5] Zabarev A Y and Iskakova L Y 2003 Colloid J. 65 159
[6] Minagawa K, Watanabe T, Koyama K and Sasaki M. 1994 Langmuir 10 3926
[7] Rwei S P, Lee H Y, Yoo S D, Wang L Y and Lin J G, 2005 Colloid Polym. Sci. 283 1253
[8] Fang F F, Jang I B and Choi H J 2007 Diamond & Related Mater. 16 1167
[9] Kim J H, Fang F F and Choi H J and Seo Y 2008 Mater. Lett. 62 2897
[10] Chen C T and Chen Y C 2005 Anal. Chem. 77 5912