Impact of Corrosive Substances on the Mucosa of the Digestive Tract

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Caustic ingestion is a major health concern in both developed and developing countries, that may lead to serious esophageal and gastric injury. Initially, the ingestion of a caustic substance produces post-caustic esophagitis, this being the acute phase, which in time by chronicization and by the methods of tissue remodeling, by fibrosis determines the post-caustic esophageal stenosis.

Key words: caustic ingestion, esophageal lesions, gastric injury

Ingestion of caustic substances often leads to severe morbidity and, frequently, death. Eymologic, the term corrosive is used for the action of acidic substances on the tissues and for the effect of the alkaline substances the term caustic is used. Currently, for both types of substances the term caustic is used [1].

For a long time, many authors agreed that the acids cause more damage to the gastric mucosa, whereas alkalis have a tendency for greater damage to the esophageal mucosa [2].

However, it is now known that these agents cause mucosal injuries in both the organs with no selective preference [3,4].

The symptomatology differs depending on the digestive segment predominantly affected. Thus, if the patient has dysphonia, dyspnoea then the lesion is in the epiglottis and larynx and represents a medical emergency that endangers the patient's life, causing acute respiratory failure [3]. If the corrosive substance has particularly damaged the esophagus, then the main symptoms of the patient are dysphagia and odynophagia. Stomach involvement causes epigastric pain or hematemesis [5,6].

Caustic agents can be acidic or alkaline in nature. Common alkali-containing caustic agents are household bleaches, drain openers, toilet bowel cleaners, dishwashing agents and detergents. Strong alkaline sold in liquid and granular forms are the principal causes of severe damage. Household bleach, dishwasher detergents, and other cleaning agents, all of which are mildly alkaline, are the most common caustic substances ingested. Alkaline commonly used are hydroxides of sodium and potassium. Acidic agents, frequently responsible, are the ones that are commonly available such as cleaning agents (hydrochloric acid) and a gold solvent, which is a mixture of hydrochloric and nitric acid (aqua regia) [7].

Alkaline substances are tasteless, colorless, odorless and have a viscous consistency, so their ingestion tends to be mostly accidental and in large quantities, [8] The most commonly used alkaline substance is sodium hydroxide.

The acids have a pungent odor and an unpleasant taste, which is why the amount of the substance ingested is small and most often suicidal. The slightly alkaline pH of the saliva buffers the acidity of the substance, thus the lesions being milder in the esophagus. [1].

Acid-containing agents implicated in caustic ingestion include toilet bowl cleaners, anti-rust compound, swimming pool cleaners, vinegar, formic acid used in the rubber tanning industry and other similar acids [9,10].

The type of caustic agent most commonly implicated in poisonings varies from country to country. In a report, the most commonly implicated caustic agent was the alkali-sodium hypochlorite, which was found in bleaches, toilet bowl cleaners, drain cleaners and household disinfectants. Local experiences from some countries - Denmark, Israel, United Kingdom, Peru, Spain, Australia, Saudi Arabia and Turkey also showed that alkaline agents were more commonly involved in caustic injury[10].

Indian data, on the other hand, showed that majority of ingestions in their country were due to acids since these were cheaper and more readily available[9,10].

Some authors reported that bleach seems to constitute approximately 30% of all caustic substances ingested by children; drain or grease cleaners make up 30% of the total and various acid or alkali cleaners and detergents such as oven, toilet, tile, and drain cleaners, lye solvents, and battery acids make up 40% [11–13].

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Such caustic substances are particularly dangerous because they are transparent (like water) and are sold as generic grease cleaners in open transparent bottles (especially 500 ml water containers) that can be easily accessed by children in developing countries [14].

Overall, the corrosive injuries of the upper GI tract are more often seen in the developing countries [15,16].

According to the American Association of Poison Control (AAPCC) since 2000, approximately 200,000 cases of ingestion of substances used in cleaning have been reported [17].

The literature describes 2 age groups prone to ingestion of caustic substances. In the first group, children between the ages of 1-5 years are framed, which are prone to accidental ingestion, compared to the second group, in which adolescents and adults who are prone to ingesting causal substances for suicidal purpose are framed. [18]

Corrosive injuries of the GI tract are seen in accidental or suicidal intake of acids or alkalis. Accidental intake is more commonly seen in children, whereas the suicidal intent is the usual cause in adults [5,10,19].

Almost all ingestions by children are accidental; the intention is usually to drink water (82%) [14,20]. However, it is difficult to accept that caustic ingestion accidents in developing countries are truly accidental, as they in fact reflect extreme negligence. In addition, caustic substances are sometimes accidentally given to babies by their mothers, grandmothers, or babysitters [14].

Most parents have low incomes and are poorly educated (mother, 88–96%; father, 62–80%) [14]. A study found that children ingesting caustic substances were more impulsive than others; hyperactivity was also recognized as a risk factor in children younger than 5 years [21].

In adults caustic ingestion is usually a deliberate suicide attempt [22]; this attempt is rare (0–0.6%) in children [14,20,21].

Another studies have shown that caustic ingestions occur more often in boys (57–64%), in children of a mean age of 3 years, mostly in summer (38.1%) and spring (31.7%), mostly in the family home, and usually in the kitchen (51.5%) [23,21].

Most caustic substances were ingested in the liquid form and events commonly occurred at home [10]. Some authors noted that, a considerable number (22–23%) of accidentally caustic ingestion occur during family visits (thus in the homes of grandparents and relatives) [20, 24].

The caustic substances are mostly liquids sold in nonoriginal containers (69–93%), particularly transparent water bottles (38%), which are often 500 ml bottles or plastic carbonated-drink bottles (23%) lacking both warning labels (72–83%) and childhood safety caps (83–93%) [14, 20].

Pathology

Caustic agents can damage any segment of the gastrointestinal tract. However, the most commonly lesions of the upper gastrointestinal tract are in the oropharynx, larynx, esophagus, and stomach [25].

However, the injuries are usually limited to the oro-oesophageal mucosa, and do not feature massive necrosis or stricture development. The physical form and especially the pH of caustic substances are important for the oesophageal damage; pH values more than 12 and less than 1.5 are associated with massive caustic injuries [8].

The severity of the injuries caused by the corrosive substances is due to the accumulation of factors: the concentration of the corrosive substance, the pH of the agent, the quantity of the ingested substance and the duration of the contact between the caustic substance and the affected tissues [5,6]. The oesophagus is the organ most commonly affected after caustic ingestion. Injuries due to caustics can be restricted to the oesophagus or stomach. Coexistent oesophageal and gastric injuries are seen in 20%–62.5% cases [26,27].

Crystalline drain cleaners (including concentrated sodium hydroxide) are strong alkalis and especially affect the proximal oesophagus, injury to which is extremely dangerous. Highly concentrated, caustic alkaline liquids usually pass quickly through the oropharynx and cause extremely severe injuries to the oesophagus [28,29].

In contrast, strong acids are bitter and are commonly expectorated. However, if swallowed, such acids quickly pass through the oesophagus and cause enormous damage to the gastric antrum, particularly if the stomach is empty. The duodenum and intestine are protected, to some extent, by pylorospasm [30]. The mucosa is damaged within seconds of contact with strong alkali or acid. Injuries caused by alkaline and acidic agents differ greatly [31].

The mechanism of acidic and alkaline injury is coagulation necrosis and liquefactive necrosis, respectively. Acid ingestion triggers coagulation necrosis of the mucosa, and formation of hard eschars usually associated with limited penetration into the mucosa [32].

In contrast, ingested alkali penetrates tissue and induces liquefactive necrosis, followed by destruction of the epithelium and submucosa and commonly extension into and sometimes through all muscle layers. Ischemia and thrombosis are early features. Then, a friable discolored eschar develops, under which tissue damage continues until the alkali is neutralized. Injuries to the oesophagus are evident principally at the three zones of constriction: the criopharyngeal area (in the upper region); the mid-oesophagus at the point of crossing of the aortic arch and the left main bronchus (in the middle region); and immediately above the oesophageo-gastric junction (in the lower region) [4].

Hemorrhage, thrombosis, and marked inflammation with edema develop in the first 24 h after injury. Depending on
the extent of burning, the inflammation may extend through the muscle layer until perforation occurs. If the injury is transmural, the necrosis may extend to the surrounding mediastinum, initiated mediastinitis. If the injury extends anteriorly, trachea-oesophageal or even aorto-oesophageal fistulae may appear. After 48 h, submucosal vessels develop thromboses, with local necrosis and gangrene. Bacterial contamination leads to the development of small intramural abscesses, which may extend to the mediastinum of full-thickness injuries. Several days later, the necrotic tissue sloughs, the edema is reduced, and neovascularization begins. Collagen deposits are seen beyond 2 weeks. This early reparative (or subacute) phase develops from the end of the first week and throughout the second week after injury. Scar formation begins in the third week, when fibroblasts proliferate to replace the submucosa and the muscularis, and stricture formation starts.

Then, mucosal re-epithelialization develops, and is usually complete by the sixth week. Adhesions may form during this period, narrowing orobliterating the oesophageal lumen. The end result may be a fibrotic stricture and a shortened oesophagus, induce gastro-oesophageal reflux and a motility disorder [20, 33].

The esophageal layer involved in fibrosis is determined by the depth of injury caused by the caustic substance. Oesophageal dysmotility may persist for several weeks, or may even become permanent if muscle is replaced by fibrous tissue [34]. Clinically significant gastric injury is relatively uncommon in children. Gastric outlet obstruction and pyloric stenosis are the most frequently gastric complications reported after caustic ingestion [35]. Gastric perforation is very rare. The transit time from the mouth to the stomach of acids is shorter than that of alkanis. Thus, although alkalis cause more severe esophageal injuries, acidic substances are associated with gastric injuries. The severity of gastric injury is determined by the amount and pH of the caustic substance ingested, and stomach fullness at the time of ingestion [36, 37].

Conclusions

Caustic ingestion remains a major health problem, especially in developing countries, where legislation is not very strict regarding the handling of caustic substances. The incidence of these accidents continues to be unacceptably high if the caustic substances were to be sold only in the original containers. It is important to reveal the need for preventive education programs for children and adults.

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