Abstract

Topological indices derived from line graph are in use since 1980’s and are mostly used in deriving Quantitative Structure-Property Relationships. Never before weighted line graphs were used for this purpose. A descriptor, $L_2$, derived from weighted line graph has been used in modeling Quantitative Structure-Activity Relationship (QSAR) for the narcotic activity of 123 diverse organic compounds. The competence of the model derived was evaluated and it was found to be more effective than the models proposed earlier. An optimized regression model was proposed by using successive exclusion of variable technique.

\[
\log \left( \frac{1}{C_{\text{50}}} \right) = 0.317 + 1.07 (\pm 0.13) R_2 - 0.55 (\pm 0.13) \sum \beta^{HI}_2 -2.50 (\pm 0.15) \sum \beta^{III}_2 + 3.56 (\pm 0.12) \ \text{Vx} - 0.04 (\pm 0.01) \ \text{L2}
\]

Keywords

QSAR, Weighted line graph, Topological indices, Aquatic toxicity

Introduction

Day by day the environment is contaminated with numerous new chemicals as a consequence of new industrial or natural biological processes [1]. Many of these exhibit adverse environmental effects and may cause severe pollution problems [2,3]. Therefore the toxicity estimation of each and every chemical is a must prior to its practical application. Performing a toxicological experiment for a given substance is expensive, time consuming and need animal testing [4]. At this end, European Union REACH (Registration, Evaluation, Authorisation and Restriction of Chemicals) legislation aimed to provide toxicity information for all new and existing chemicals [5] and promotes for the use of sufficiently validated computational prediction models based on QSAR as an alternative to fill in the toxicity data gaps [4,6,7]. Various regulatory agencies like the United States Environmental Protection Agency (US EPA), European Centre for the Validation of Alternative Methods (ECVAM) of the European Union, the Agency for Toxic Substances and Disease Registry (ATSDR) and the European Union Commission’s Scientific Committee on Toxicity, Ecotoxicity and Environment (CSTEE), etc. have recommended the use of QSAR models for the risk assessment of chemical compounds [8].

A QSAR model to be valid within REACH is to be obtained in accordance with the five principles set by Organization for Economic Cooperation and Development (OECD) [9]. These models are mostly species based, and predict the environmental toxicity of contaminants taking toxicity as a function of molecular structure or physicochemical properties of the toxicant [4,10-19]. QSAR models were developed and reported on a congeneric compound (i.e. chemical classes) or by considering a chemical’s mode (and mechanism) of action [10,20-27]. However, a global QSAR model is still in need.

Aquatic toxicity is one of the most important measures in the ecotoxicological risk assessment of chemicals, as it constitutes a chain of different trophic level species for toxicity assays and in addition, it is less cumbersome than other test methods [28]. Due to the habitat characteristics, amphibians are often the main vertebrate group at risk of exposure to contaminants in both terrestrial and aquatic environment [29]. However, the limited data pertaining to the effects of these contaminants on amphibians entice on the QSAR studies to propose a structural model for optimized activity [30]. Frogs are common amphibians bridging aquatic organisms and terrestrial animals. The tadpoles are proved to be more sensitive to hazards than adult frogs [31] and have been recommended by the EU-TGD [32] for narcotic analysis.

*Corresponding author: Chandana Adhikari, Centurion University of Technology and Management, Odisha, India

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QSAR studies generally stem on reliable data of a wide range of compounds, which is mostly lacking in chemical literature. In this context a huge data on narcotic property of organic compounds were collected by Overton [33,34]. Abraham and Rafols have extended the data set of Overton by adding new compounds in terms of their narcotic behavior from various sources making it a data set of 123 diverse organic compounds [35]. With this extensive data set and using the linear free energy relationship (LEFR), as is proposed by many authors [31,36-38] for narcosis phenomenon, they expressed the tadpole narcosis as a linear function of solvato-metric parameters (Eq. 1).

$$\log \left( \frac{1}{C_{\text{nar}}} \right) = R_2 \left( \sum_{i=1}^{n} \pi^H_i, \sum_{i=1}^{n} \beta^H_i, V_x \right)$$

Where the descriptors used are: $R_2$, the solute excess molar refraction, $\sum_{i=1}^{n} \pi^H_i$ the solute dipolarity/polarizability, $\sum_{i=1}^{n} \beta^H_i$ the solute hydrogen-bond basicity and $V_x$ is solute volume. Excluding nine outliers, the regression model has a correlation coefficient of $r = 0.952$ and standard deviation (SD) = 0.341.

Applying Overton Meyer’s relationship [39] for tadpole narcosis, Abraham and Rafols derived the correlation equation for narcotic property in terms of water-octanol partition coefficient ($\log P_{\text{oct}}$) with correlation coefficient $r = 0.9301$ and standard deviation (SD) = 0.414. That indicates water-octanol partition coefficient alone is sufficient to explain the tadpole narcosis as a linear function of solvato-metric parameters.

The objective of the present work is to introduce a new molecular descriptor, weighted line graph index (L2) in the QSAR world. Hansch, et al. put the milestone in the way to search parameters as a substitute to log P [46]. Probably this forms the basis to use topological and/or other molecular descriptor in place of log P in the QSAR for toxicity of chemical compounds as is done by Agarwal, et al. [47-53].

In this context we have taken tadpole narcosis as a general data set and tried to express the narcotic activity in terms of topological indices. While doing so, they combined the distance based topological indices ($W$, $Sz$, $\chi^H$ (B), $J$ and log RB) with Abraham’s molecular descriptors and gave a hexa-parametric regression model for tadpole narcosis with better statistical significance ($r = 0.9592$, $SE = 0.3217$) [54] than Abraham and Rafols model [35].

In a subsequent study, Jaiswal and Khadikar examined the potential of distance based topological indices Wiener (W), PI (Padmakan-Ivan), Randic connectivity ($^1x^1$, $^1x^2$) and Kier and Hall valence connectivity ($^0x_v$, $^1x_v$, $^2x_v$) over log P in modeling tadpole narcosis by taking a heterogeneous set of 22 compounds [55]. The best model is obtained in terms of first order valance connectivity index $^1x_v$ with $r = 0.9806$ as presented in Eq. 2.

$$\log \left( \frac{1}{C_{\text{nar}}} \right) = -0.7839 (\pm 0.1289) + 1.1859 (\pm 0.0544) ^1x_v$$

The positive coefficient of $^1x_v$ in the above equation shows that the degree of unsaturation and the presence of heteroatom enhance the tadpole narcosis as the descriptor distinguishes the degree of unsaturation and the presence of heteroatom in the considered organic compound.

A hydrogen depleted molecular graph (G) can be obtained by taking the atoms as vertices and the bond between the atoms as the edges. In a line graph (L(G)) of G, the vertices correspond to the edges of G and two vertices of the L(G) are adjacent if their corresponding edges of the molecular graph G have a common vertex. The prologue of line graph indices in the QSPR world is due to Bertz [56]. A sporadic report of the use of line graphs in QSPR was found in literature for some mutually unrelated chemical fields [57-65].

Herein, we derived a novel weighted line graph index (L2) from the weighted iterated line graph of hydrogen depleted molecular graphs of 123 organic compounds with narcotic activity. So far, parameters derived from line graph have not yet been used in QSAR studies. Line graph has been used for the carboneous compounds while for heteroatomic compounds the weighted line graph will be more appropriate for derivation of molecular descriptors.

**Enumeration of Weighted Line Graph Index (L2)**

A weighted line graph is the line graph $L_w (G)$ of a molecular graph, G, in which each vertex of $L (G)$ is assigned with a nonnegative number $w (u_i)$, referred to the weight of $u_i$. Starting with the hydrogen depleted molecular graph G or the zeroth order line graph L0 (G), each vertex of the L (G) is assigned with a value equal to the weight on the connecting edge of (G) attached to the vertex ‘v’ to obtain zero order weighted line graph $L_w (G)^0$. The weight of the edge of G is the weight of the vertex $w (u_i)$ is given by

$$v_{i-j} = w (u_i) \cdot \frac{1}{\sqrt{\delta_i \delta_j}}$$

Where $\delta_i$ and $\delta_j$ are the weight of the vertices $v_i$ and $v_j$ respectively.

$$\delta = z^x - H$$

Where, $z^x$ and H are the number of valence electron and H-atom bonded to the ith atom respectively. While tagging weight to the multiple bonds, the weight of the edge is multiplied by a factor equal to the multiplicity of the concerned bond.

The molecular descriptor for the zeroth order weighted line graph is defined as

$$L0 = \sum_{edges} v_{i-j} = \sum_{edges} \frac{1}{\sqrt{\delta_i \delta_j}}$$

In a similar manner, by considering the edge of $L_w (G)^0$ as the vertex and their connectivity, the first iterated line graph, $L_w (G)^1$, is constructed and the sum of the weight of the edges of the line graph is considered as the molecular descriptor L1. Accordingly, higher iterated graphs $L_w (G)^n$ and corresponding molecular descriptors $L_i$ can be determined. An example of weighted line graphs of ethylisobutanoate is presented in Figure 1.
Ethylisobutanoate

\[ L_0 = \sum_{\text{edges}} v_{i,j} = 3.0514 \]

\[ L_0 (G). \]

\[ L_1 = \sum_{\text{edges}} v_{i,j} = 23.4605 \]

\[ L_1 (G) \]

\[ L_2 = \sum_{\text{edges}} v_{i,j} = 4.4638 \]

Iterated Line Graph ILG of ethylisobutanoate, \( L_2 (G) \)

**Figure 1:** Representations of weighted line graphs \( L_w(G)_0, L_w(G)_1 \) and \( L_w(G)_2 \) of ethylisobutanoate.
Table 1: The narcotic activity in term minimum concentration required for narcosis of different compounds and their molecular descriptors 
\( (R_2, \sum \pi^H_2, \sum \beta^0_2, V_x) \) along with log \( P_{\text{act}} \) value\(^*\).

| Sl.No | Compound            | log (1/C\(_{\text{nar}}\)) | \( R_2 \) | \( \sum \pi^H_2 \) | \( \sum \alpha^H_2 \) | \( \sum \beta^0_2 \) | \( V_x \) | log \( P_{\text{act}} \) |
|-------|---------------------|--------------------------|----------|-----------------|-----------------|-----------------|--------|-----------------|
| 1     | Pentane             | 2.55                     | 0.00     | 0.00            | 0.00            | 0.00            | 0.81   | 3.39            |
| 2     | 2-Methylbut-2-ene   | 2.64                     | 0.16     | 0.08            | 0.00            | 0.07            | 0.77   | 2.67            |
| 3     | Trichloromethane    | 2.85                     | 0.43     | 0.49            | 0.15            | 0.02            | 0.62   | 1.97            |
| 4     | Tetrachloromethane  | 3.14                     | 0.46     | 0.38            | 0.00            | 0.00            | 0.74   | 2.83            |
| 5     | Chloroethane        | 2.35                     | 0.23     | 0.40            | 0.00            | 0.10            | 0.51   | 1.43            |
| 6     | 1,2-Dichloroethane  | 2.63                     | 0.42     | 0.64            | 0.10            | 0.11            | 0.64   | 1.48            |
| 7     | Bromoethane         | 2.57                     | 0.37     | 0.40            | 0.00            | 0.12            | 0.57   | 1.61            |
| 8     | Iodoethane          | 2.96                     | 0.64     | 0.40            | 0.00            | 0.15            | 0.65   | 2.00            |
| 9     | Diethyl ether       | 1.47                     | 0.04     | 0.25            | 0.00            | 0.45            | 0.73   | 0.89            |
| 10    | Propanone           | 0.54                     | 0.18     | 0.70            | 0.04            | 0.49            | 0.55   | -0.24           |
| 11    | Butanone            | 1.04                     | 0.17     | 0.70            | 0.00            | 0.51            | 0.69   | 0.29            |
| 12    | Pentan-2-one        | 1.72                     | 0.14     | 0.68            | 0.00            | 0.51            | 0.83   | 0.84            |
| 13    | Pentan-3-one        | 1.54                     | 0.15     | 0.66            | 0.00            | 0.51            | 0.83   | 0.82            |
| 14    | Camphor             | 2.88                     | 0.45     | 0.85            | 0.00            | 0.56            | 1.32   | 2.53            |
| 15    | Ethylformate        | 1.16                     | 0.15     | 0.66            | 0.00            | 0.38            | 0.61   | 0.27            |
| 16    | Methyl acetate      | 1.10                     | 0.14     | 0.64            | 0.00            | 0.45            | 0.61   | 0.18            |
| 17    | Ethyl acetate       | 1.52                     | 0.11     | 0.62            | 0.00            | 0.45            | 0.75   | 0.73            |
| 18    | Propyl acetate      | 1.96                     | 0.09     | 0.60            | 0.00            | 0.45            | 0.89   | 1.24            |
| 19    | Butyl acetate       | 2.30                     | 0.07     | 0.60            | 0.00            | 0.45            | 1.03   | 1.82            |
| 20    | Isobutyl acetate    | 2.24                     | 0.05     | 0.57            | 0.00            | 0.47            | 1.03   | 1.60            |
| 21    | Pentyl acetate      | 2.72                     | 0.07     | 0.60            | 0.00            | 0.45            | 1.17   | 2.01            |
| 22    | Ethylpropanoate     | 1.96                     | 0.09     | 0.58            | 0.00            | 0.45            | 0.89   | 1.21            |
| 23    | Ethylbutanoate      | 2.37                     | 0.07     | 0.58            | 0.00            | 0.45            | 1.03   | 1.73            |
| 24    | Ethyl pentanoate    | 2.72                     | 0.05     | 0.58            | 0.00            | 0.45            | 1.17   | 2.26            |
| 25    | Butyl pentoanoate   | 3.60                     | 0.03     | 0.56            | 0.00            | 0.45            | 1.45   | 3.32            |
| 26    | Ethylisobutanoate   | 2.24                     | 0.03     | 0.55            | 0.00            | 0.47            | 1.03   | 1.51            |
| 27    | Triacetin           | 1.64                     | 0.14     | 1.30            | 0.00            | 0.35            | 1.60   | 0.25            |
| 28    | Acetonitrile        | 0.44                     | 0.24     | 0.90            | 0.07            | 0.32            | 0.40   | -0.34           |
| 29    | Nitromethane        | 1.09                     | 0.31     | 0.95            | 0.06            | 0.31            | 0.42   | -0.33           |
| 30    | Pentanamide         | 1.30                     | 0.40     | 1.30            | 0.50            | 0.62            | 0.93   | 0.35            |
| 31    | N-Ethylurethane     | 1.40                     | 0.24     | 0.82            | 0.24            | 0.61            | 0.99   | 1.02            |
| 32    | Methanol            | 0.24                     | 0.28     | 0.44            | 0.43            | 0.47            | 0.31   | -0.74           |
| 33    | Ethanol             | 0.54                     | 0.25     | 0.42            | 0.37            | 0.48            | 0.45   | -0.30           |
| 34    | Propan-1-ol         | 0.96                     | 0.24     | 0.42            | 0.37            | 0.48            | 0.59   | 0.25            |
| 35    | Propan-2-ol         | 0.89                     | 0.21     | 0.36            | 0.33            | 0.56            | 0.59   | 0.05            |
| 36    | Butan-1-ol          | 1.42                     | 0.22     | 0.42            | 0.37            | 0.48            | 0.73   | 0.84            |
| 37    | 2-Methyl propan-1-ol| 1.35                     | 0.22     | 0.39            | 0.37            | 0.48            | 0.73   | 0.76            |
| 38    | 2-Methyl propan-2-ol| 0.89                     | 0.18     | 0.30            | 0.31            | 0.60            | 0.73   | 0.35            |
| 39    | 3-Methyl butan-1-ol | 1.64                     | 0.19     | 0.39            | 0.37            | 0.48            | 0.87   | 1.28            |
| 40    | 2-Methyl butan-2-ol | 1.20                     | 0.19     | 0.30            | 0.31            | 0.60            | 0.87   | 0.89            |
| 41    | Octan-1-ol          | 3.40                     | 0.20     | 0.42            | 0.37            | 0.48            | 1.30   | 3.07            |
|   | Chemical        | 42  | 43  | 44  | 45  | 46  | 47  | 48  | 49  | 50  | 51  | 52  | 53  | 54  | 55  | 56  | 57  | 58  | 59  | 60  | 61  | 62  | 63  | 64  | 65  | 66  | 67  | 68  | 69  | 70  | 71  | 72  | 73  | 74  | 75  | 76  | 77  | 78  | 79  | 80  | 81  | 82  | 83  | 84  | 85  | 86  |
|---|----------------|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
|   |                | 3.97| 0.40| 0.48| 0.32| 0.61| 1.47| 3.31| 0.19| 0.40| 0.90| 0.58| 0.78| 0.51| -1.36| 2.09| 0.39| 0.35| 0.00| 0.24| 0.55| 1.18| 3.28| 0.88| 0.26| 0.00| 0.03| 0.49| 1.94| 1.96| 0.00| 1.00| 0.00| 1.06| 1.39| 0.80| 2.68| 0.61| 0.52| 0.00| 0.14| 0.72| 2.13| 3.42| 0.62| 0.52| 0.00| 0.16| 1.00| 3.20| 4.19| 1.34| 0.92| 0.00| 0.20| 1.09| 3.30| 5.25| 2.06| 1.29| 0.00| 0.26| 1.45| 4.46| 2.82| 0.71| 0.75| 0.00| 0.29| 0.92| 2.11| 3.35| 0.82| 1.01| 0.00| 0.45| 1.12| 2.21| 3.05| 0.81| 1.00| 0.00| 0.50| 1.12| 2.03| 1.96| 0.96| 0.96| 0.26| 0.50| 0.82| 0.90| 2.85| 0.96| 0.84| 0.00| 0.47| 1.10| 2.31| 4.43| 0.70| 0.88| 0.60| 0.38| 1.42| 3.50| 4.74| 0.68| 1.20| 0.00| 0.44| 1.48| 3.82| 2.31| 0.87| 1.40| 0.50| 0.67| 1.11| 1.16| 2.09| 0.97| 1.63| 0.48| 0.86| 1.31| 1.05| 2.55| 0.94| 1.60| 0.48| 0.84| 1.45| 1.58| 2.28| 0.81| 0.89| 0.60| 0.30| 0.78| 1.50| 2.92| 0.84| 0.86| 0.52| 0.30| 0.92| 1.98| 2.75| 0.82| 0.88| 0.57| 0.34| 0.92| 1.98| 2.75| 0.82| 0.87| 0.57| 0.31| 0.92| 1.97| 4.26| 0.82| 0.79| 0.52| 0.44| 1.34| 3.30| 4.52| 0.81| 0.89| 0.56| 0.41| 1.48| 3.83| 2.57| 0.84| 0.91| 0.22| 0.52| 0.97| 1.32| 2.12| 0.97| 1.07| 0.85| 0.52| 0.83| 0.88| 1.64| 0.98| 1.00| 1.10| 0.58| 0.83| 0.80| 2.12| 1.00| 1.00| 1.16| 0.60| 0.83| 0.59| 2.48| 1.04| 1.04| 0.32| 0.67| 1.13| 1.21| 3.91| 0.95| 0.99| 0.22| 0.51| 1.35| 2.99| 2.18| 1.25| 1.72| 0.49| 0.78| 1.18| 0.73| 3.24| 1.06| 1.79| 0.00| 0.46| 1.06| 1.32| 2.37| 0.95| 1.90| 0.00| 0.46| 0.96| 0.80| 2.78| 0.99| 1.60| 0.00| 0.52| 1.02| 1.05| 1.60| 0.63| 0.84| 0.00| 0.47| 0.68| 0.65| 2.72| 1.27| 0.97| 0.00| 0.51| 1.04| 2.03| 1.89| 1.32| 1.50| 0.00| 1.48| 1.55| 0.23| 1.92| 1.40| 1.55| 0.00| 1.34| 1.36| -0.07| 2.76| 2.20| 2.34| 0.86| 1.79| 2.06| 0.76| 2.34| 1.11| 1.40| 0.77| 0.77| 1.07| 0.83| 0.77| 0.46| 1.30| 0.54| 0.68| 0.51| -1.26| 0.57| 0.26| 0.82| 0.24| 0.61| 0.85| 0.34| 3.51| 0.87| 1.34| 0.00| 0.94| 1.37| 1.17|   |
| Compound                | Log 1/C<sub>nar</sub> | W 0 | W 1 | W 2 | W 3 | log RB | J   | Log 1/C<sub>nar</sub> |
|-------------------------|-----------------------|-----|-----|-----|-----|--------|-----|-----------------------|
| 2-Propylpiperidine      | 3.48                  | 0.36| 0.44| 0.10| 0.69| 1.23   | 2.15|
| Urea                    | 0.60                  | 0.50| 1.00| 0.50| 0.90| 0.46   | -2.11|
| N-Ethylurethane         | 1.46                  | 0.24| 0.82| 0.24| 0.61| 0.99   | 1.02|
| N-Propylurethane        | 2.18                  | 0.23| 0.82| 0.24| 0.61| 1.13   | 1.55|
| N-Isobutylurethane      | 2.50                  | 0.20| 0.79| 0.24| 0.61| 1.27   | 2.02|
| N-Isopentylurethane     | 2.93                  | 0.19| 0.79| 0.24| 0.61| 1.41   | 2.64|
| Methanol                | 0.23                  | 0.28| 0.44| 0.43| 0.47| 0.31   | -0.74|
| Ethanol                 | 0.72                  | 0.25| 0.42| 0.37| 0.48| 0.45   | -0.30|
| Propan-1-ol             | 1.14                  | 0.24| 0.42| 0.37| 0.48| 0.59   | 0.25|
| Butan-1-ol              | 1.97                  | 0.22| 0.42| 0.37| 0.48| 0.73   | 0.84|
| Pentan-1-ol             | 2.54                  | 0.22| 0.42| 0.37| 0.48| 0.87   | 1.51|
| Hexan-1-ol              | 3.24                  | 0.21| 0.42| 0.37| 0.48| 0.91   | 2.03|
| Heptan-1-ol             | 3.64                  | 0.21| 0.42| 0.37| 0.48| 1.15   | 2.62|
| Octan-1-ol              | 4.24                  | 0.20| 0.42| 0.37| 0.48| 1.30   | 3.07|
| Nonan-1-ol              | 4.43                  | 0.19| 0.42| 0.37| 0.48| 1.44   | 3.77|
| Decan-1-ol              | 1.90                  | 0.19| 0.42| 0.37| 0.48| 1.58   | 4.18|
| Undecan-1-ol            | 5.09                  | 0.18| 0.42| 0.37| 0.48| 1.72   | 4.72|
| Dodecan-1-ol            | 5.33                  | 0.18| 0.42| 0.37| 0.48| 1.86   | 5.13|
| Butan-2-ol              | 1.77                  | 0.22| 0.36| 0.33| 0.56| 0.73   | 0.65|
| Pentan-2-ol             | 2.32                  | 0.20| 0.36| 0.33| 0.56| 0.87   | 1.25|
| Hexan-2-ol              | 2.86                  | 0.19| 0.36| 0.33| 0.56| 1.01   | 1.76|
| Heptan-2-ol             | 3.48                  | 0.19| 0.36| 0.33| 0.56| 1.15   | 2.31|
| Octan-2-ol              | 4.21                  | 0.16| 0.36| 0.33| 0.56| 1.30   | 2.90|
| Cyclohexanol            | 2.30                  | 0.46| 0.54| 0.32| 0.57| 0.90   | 1.23|
| Cycloheptanol           | 2.89                  | 0.51| 0.54| 0.32| 0.58| 1.05   | 1.80|
| Cyclooctanol            | 3.41                  | 0.58| 0.54| 0.32| 0.58| 1.19   | 2.36|
| Cyclodecanol            | 4.08                  | 0.62| 0.54| 0.32| 0.58| 1.47   | 3.48|
| Pentan-1,5-diol         | 1.72                  | 0.39| 0.95| 0.72| 0.92| 0.93   | -0.25|
| Hexan-1,6-diol          | 1.60                  | 0.39| 0.95| 0.75| 0.92| 1.07   | 0.25|
| Heptan-1,7-diol         | 2.48                  | 0.38| 0.95| 0.75| 0.92| 1.21   | 0.78|
| Octan-1,8-diol          | 3.02                  | 0.38| 0.95| 0.75| 0.92| 1.35   | 1.32|
| Nonan-1,9-diol          | 3.19                  | 0.37| 0.95| 0.75| 0.92| 1.49   | 1.85|
| Decan-1,10-diol         | 3.60                  | 0.37| 0.95| 0.75| 0.92| 1.64   | 2.39|
| Dodecan-1,12-diol       | 4.41                  | 0.36| 0.95| 0.75| 0.92| 1.92   | 3.46|
| Acetal                  | 1.92                  | 0.00| 0.67| 0.00| 0.08| 1.07   | 0.84|
| Benzamide               | 2.52                  | 0.99| 1.50| 0.49| 0.67| 0.97   | 0.64|
| Benzylalcohol           | 2.70                  | 0.80| 0.87| 0.39| 0.56| 0.92   | 1.10|

*All the values were taken from [35].

**Database and Methodology**

The narcotic activity (log 1/C<sub>nar</sub>) along with the Abraham’s molecular descriptors (R<sub>2</sub>, ∑π<sup>α</sup>/<sub>2</sub>, ∑β<sup>α</sup>/<sub>2</sub>, V<sub>1</sub>) and logP<sub>oct</sub> for 123 compounds collected from Abraham and Rafols’s study [35] are presented in Table 1. The narcosis values vary from 0.19 for ethane-1, 2-diol to 5.33 for dodecanol. The topological indices (W, Sz, ℙ (B), J and log RB) for the same set of 123 compounds were collected from Ref. [54] and presented in Table 2. The weighted line graph parameters L0, L1 and L2 were derived and L2 was selected for the present QSAR studies due to its nondegeneracy. Further, due to higher complexity in deriving Li of higher orders, only L2 was considered in the present study. The L2 were calculated by above reported method and presented in Table 2, along with...
Table 2: Molecular descriptors showing degeneracy.

| Compound | Sz  | W  | $\chi (=B)$ | J  | log(RB) | L2   |
|----------|-----|----|-------------|----|---------|------|
| 1        | 20  | 20 | 2.41        | 2.19 | 5.66    | 1.09 |
| 2        | 20  | 20 | 2.41        | 2.19 | 5.66    | 2.17 |
| 3        | 9   | 9  | 1.73        | 2.32 | 2.08    | 0.65 |
| 4        | 16  | 16 | 2.00        | 3.02 | 4.16    | 2.27 |
| 5        | 4   | 4  | 1.41        | 1.63 | 0.69    | 0.00 |
| 6        | 10  | 10 | 1.91        | 1.97 | 2.48    | 0.37 |
| 7        | 4   | 4  | 1.41        | 1.63 | 0.69    | 0.00 |
| 8        | 4   | 4  | 1.41        | 1.63 | 0.69    | 0.00 |
| 9        | 20  | 20 | 2.41        | 2.19 | 5.66    | 0.72 |
| 10       | 9   | 9  | 1.73        | 2.32 | 2.08    | 1.40 |
| 11       | 18  | 18 | 2.27        | 2.54 | 4.97    | 2.14 |
| 12       | 32  | 32 | 2.77        | 2.63 | 9.53    | 2.57 |
| 13       | 32  | 32 | 2.77        | 2.63 | 9.53    | 2.83 |
| 14       | 174 | 123| 4.98        | 2.40 | 39.59   | 20.97|
| 15       | 20  | 20 | 2.41        | 2.19 | 5.66    | 0.64 |
| 16       | 18  | 18 | 2.27        | 2.54 | 4.97    | 1.64 |
| 17       | 32  | 32 | 2.77        | 2.63 | 9.53    | 1.92 |
| 18       | 52  | 52 | 3.27        | 2.68 | 15.93   | 1.59 |
| 19       | 79  | 79 | 3.77        | 2.72 | 24.30   | 2.09 |
| 20       | 71  | 71 | 3.66        | 3.10 | 22.29   | 4.24 |
| 21       | 114 | 114| 4.27        | 2.75 | 34.77   | 2.59 |
| 22       | 50  | 50 | 3.31        | 2.83 | 15.42   | 2.58 |
| 23       | 76  | 76 | 3.81        | 2.86 | 23.61   | 3.02 |
| 24       | 110 | 110| 4.31        | 2.88 | 33.93   | 3.52 |
| 25       | 200 | 200| 5.31        | 3.02 | 60.33   | 4.47 |
| 26       | 71  | 71 | 3.66        | 3.10 | 22.29   | 4.46 |
| 27       | 424 | 424| 6.91        | 3.84 | 128.67  | 8.26 |
| 28       | 4   | 4  | 1.41        | 1.63 | 0.69    | 0.00 |
| 29       | 9   | 9  | 1.73        | 2.32 | 2.08    | 1.17 |
| 30       | 143 | 143| 4.68        | 3.13 | 44.24   | 2.81 |
| 31       | 98  | 98 | 4.22        | 3.32 | 31.06   | 2.61 |
| 32       | 4   | 4  | 1.41        | 1.63 | 0.69    | 0.00 |
| 33       | 10  | 10 | 1.91        | 1.97 | 2.48    | 0.00 |
| 34       | 20  | 20 | 2.41        | 2.19 | 5.66    | 0.49 |
| 35       | 18  | 18 | 2.27        | 2.54 | 4.97    | 1.33 |
| 36       | 35  | 35 | 2.91        | 2.34 | 10.45   | 0.99 |
| 37       | 32  | 32 | 2.77        | 2.63 | 9.53    | 2.38 |
| 38       | 28  | 28 | 2.56        | 3.17 | 8.15    | 4.96 |
| 39       | 29  | 29 | 2.84        | 2.09 | 8.15    | 2.90 |
| 40       | 44  | 44 | 3.12        | 3.36 | 13.52   | 5.83 |
| 41       | 165 | 165| 4.91        | 2.65 | 48.96   | 2.99 |
| 42       | 291 | 191| 5.65        | 2.50 | 61.54   | 11.58|
| 43       | 35  | 35 | 2.91        | 2.34 | 10.45   | 0.40 |
|   |   |   |   |   |   |
|---|---|---|---|---|---|
| 44 | 10 | 10 | 1.91 | 1.97 | 2.48 | 0.00 |
| 45 | 4  | 4  | 1.41 | 1.63 | 0.69 | 0.00 |
| 46 | 166| 166| 5.18 | 3.79 | 53.17| 5.55 |
| 47 | 54 | 27 | 3.00 | 2.00 | 7.45 | 2.83 |
| 48 | 110| 62 | 3.79 | 2.19 | 19.00| 6.96 |
| 49 | 243| 109| 4.97 | 1.93 | 34.42| 9.42 |
| 50 | 677| 271| 6.95 | 1.48 | 86.81| 16.06|
| 51 | 109| 64 | 3.93 | 2.13 | 19.70| 4.85 |
| 52 | 192| 121| 4.86 | 2.25 | 38.34| 7.09 |
| 53 | 200| 125| 4.86 | 2.17 | 39.08| 7.11 |
| 54 | 142| 88 | 4.30 | 2.23 | 27.66| 6.93 |
| 55 | 142| 88 | 4.30 | 2.23 | 27.66| 4.57 |
| 56 | 142| 88 | 4.30 | 2.23 | 27.66| 6.75 |
| 57 | 505| 307| 6.88 | 1.82 | 95.86| 10.12|
| 58 | 498| 399| 6.93 | 1.82 | 113.75| 10.38|
| 59 | 228| 156| 5.22 | 2.32 | 49.70| 6.76 |
| 60 | 360| 250| 6.15 | 2.41 | 79.22| 8.90 |
| 61 | 445| 316| 6.65 | 2.39 | 98.64| 9.18 |
| 62 | 109| 64 | 3.93 | 2.13 | 19.70| 4.44 |
| 63 | 142| 86 | 4.34 | 2.30 | 27.15| 6.63 |
| 64 | 146| 88 | 4.33 | 2.23 | 27.66| 6.49 |
| 65 | 150| 90 | 4.33 | 2.18 | 28.07| 6.52 |
| 66 | 366| 244| 6.19 | 2.50 | 77.90| 9.13 |
| 67 | 372| 252| 6.10 | 2.40 | 78.98| 14.03|
| 68 | 184| 117| 4.88 | 2.34 | 37.42| 6.65 |
| 69 | 184| 117| 4.88 | 2.34 | 37.42| 6.13 |
| 70 | 192| 121| 4.86 | 2.25 | 38.34| 6.02 |
| 71 | 200| 125| 4.86 | 2.17 | 39.08| 6.05 |
| 72 | 303| 198| 5.81 | 2.41 | 63.20| 9.16 |
| 73 | 376| 254| 6.30 | 2.39 | 80.42| 9.72 |
| 74 | 340| 250| 6.13 | 2.40 | 79.22| 6.59 |
| 75 | 308| 144| 5.36 | 1.93 | 45.79| 10.04|
| 76 | 185| 106| 4.87 | 2.01 | 33.51| 10.00|
| 77 | 248| 147| 5.40 | 1.91 | 46.48| 10.04|
| 78 | 54 | 27 | 3.00 | 2.00 | 7.45 | 2.61 |
| 79 | 243| 109| 4.97 | 1.93 | 34.42| 9.03 |
| 80 | 414| 284| 6.70 | 2.00 | 90.72| 13.69|
| 81 | 423| 258| 6.54 | 2.23 | 84.41| 15.56|
| 82 | 2224| 940| 11.20| 1.62 | 297.68| 34.00|
| 83 | 340| 250| 6.13 | 2.40 | 79.22| 6.53 |
| 84 | 29 | 29 | 2.64 | 2.99 | 8.55 | 1.17 |
| 85 | 70 | 70 | 3.68 | 3.17 | 22.00| 1.42 |
| 86 | 323| 242| 5.93 | 1.91 | 72.37| 11.21|
| 87 | 187| 121| 4.84 | 2.25 | 38.34| 5.72 |
| 88 | 65 | 65 | 3.55 | 3.46 | 20.57| 0.97 |
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|   |   |   |   |   |
|---|---|---|---|---|
| 89 | 98 | 98 | 4.22 | 3.32 |
| 90 | 135 | 135 | 4.72 | 3.38 |
| 91 | 168 | 168 | 5.13 | 3.70 |
| 92 | 219 | 219 | 5.63 | 3.73 |
| 93 | 4 | 4 | 1.41 | 1.63 |
| 94 | 10 | 10 | 1.91 | 21.97 |
| 95 | 20 | 20 | 2.41 | 2.19 |
| 96 | 35 | 35 | 2.91 | 2.34 |
| 97 | 56 | 56 | 3.41 | 2.45 |
| 98 | 84 | 84 | 3.91 | 2.53 |
| 99 | 120 | 120 | 4.41 | 2.60 |
| 100 | 165 | 165 | 4.91 | 2.65 |
| 101 | 220 | 220 | 5.41 | 2.69 |
| 102 | 286 | 286 | 5.91 | 0.73 |
| 103 | 364 | 364 | 6.41 | 0.73 |
| 104 | 455 | 455 | 6.91 | 0.73 |
| 105 | 31 | 31 | 2.81 | 2.83 |
| 106 | 50 | 50 | 3.31 | 2.83 |
| 107 | 76 | 76 | 3.81 | 2.86 |
| 108 | 110 | 110 | 4.31 | 2.88 |
| 109 | 153 | 153 | 4.81 | 2.89 |
| 110 | 109 | 64 | 3.93 | 2.13 |
| 111 | 121 | 88 | 4.43 | 2.18 |
| 112 | 217 | 217 | 4.93 | 2.16 |
| 113 | 381 | 206 | 5.93 | 2.17 |
| 114 | 120 | 120 | 4.41 | 2.60 |
| 115 | 165 | 165 | 4.91 | 2.65 |
| 116 | 220 | 220 | 5.41 | 2.69 |
| 117 | 286 | 286 | 5.91 | 2.73 |
| 118 | 364 | 364 | 6.41 | 2.76 |
| 119 | 455 | 455 | 6.91 | 2.78 |
| 120 | 560 | 560 | 7.41 | 2.81 |
| 121 | 75 | 75 | 3.81 | 2.92 |
| 122 | 228 | 156 | 5.22 | 2.32 |
| 123 | 148 | 94 | 4.43 | 2.08 |

The other topological indices under study. Comparing L2 with the rest of the descriptors, it is found that unlike all other descriptors, L2 is nondegenerate throughout the data set and thus L2 is giving a unique numerical identity to each individual compound. The L2 value ranges from 0 to 34.004. Zero values are assigned to the molecular structures with three and less number of atoms other than hydrogen. With complexity in the chemical structure, the L2 value increases. Among the molecules reported in the present study, the maximum value is due to morphine.

The interrelationship of L2 with Abraham’s molecular descriptors \( R_2, \sum \pi_2^0, \sum \beta_2^0, V_x \) and the topological descriptors \((W, Sz, \chi(=B), J\) and \(\log RB)\) were analyzed by constructing the cross correlation matrix (Table 3). It is found that no single descriptor can explain the narcotic activity of the compounds effectively. The maximum correlation coefficient value for a single parametric equation (\(\log 1/C_{nar} = f (molecular\ descriptor)\)) is found to be 0.6652 for Vx while for L2 the value is 0.4135. Since the independent variables having high correlation coefficient in the correlation matrix represent same statistical significance, the parameters showing poor
Table 3: Cross correlation matrix of various molecular descriptors and narcotic values.

|               | log(1/C_{nar}) | R_{2} | \( \sum \alpha_{2}^{H} \) | \( \sum \beta_{2}^{o} \) | V_{x} | Sz | W | \( 1/x(=B) \) | J | log(RB) | L2 |
|---------------|----------------|-------|---------------------------|---------------------------|-------|----|---|----------------|---|--------|----|
| log(1/C_{nar})| 1              |       |                           |                           |       |    |    |                 |   |        |    |
| R_{2}         | 0.28           | 1     |                           |                           |       |    |    |                 |   |        |    |
| \( \pi_{2}^{H} \) | 0.03           | 0.72  | 1                         |                           |       |    |    |                 |   |        |    |
| \( \sum \beta_{2}^{o} \) | -0.10          | 0.30  | 0.56                       | 0.43                       | 1     |    |    |                 |   |        |    |
| V_{x}         | 0.67           | 0.27  | 0.37                       | 0.15                       | 0.45  | 1  |    |                 |   |        |    |
| Sz            | 0.41           | 0.61  | 0.60                       | 0.31                       | 0.57  | 0.70| 1  |                 |   |        |    |
| W             | 0.51           | 0.42  | 0.54                       | 0.37                       | 0.59  | 0.85| 0.91| 1               |   |        |    |
| \( 1/x(=B) \) | 0.58           | 0.52  | 0.62                       | 0.35                       | 0.59  | 0.89| 0.82| 0.91           | 1 |        |    |
| J             | -0.16          | -0.18 | -0.10                      | 0.05                       | 0.04  | -0.08| -0.10| -0.10          | 1 |        |    |
| log(RB)       | 0.49           | 0.46  | 0.57                       | 0.36                       | 0.61  | 0.85| 0.93| 0.96           | 0.92| -0.07          | 1 |
| L2            | 0.41           | 0.72  | 0.65                       | 0.09                       | 0.47  | 0.64| 0.82| 0.69           | 0.77| -0.14          | 0.73| 1 |

Correlations are only considered in the optimization process. As indicated by the correlation parameters, the molecular descriptor L2 does not have good correlation with other descriptors under consideration. Accordingly the optimization process is carried out by taking all the ten descriptors under study.

Multiple regression analysis was used to find out the QSAR for the narcotic activity of 123 diverse organic compounds. The regression model was optimized by reducing the number of variables using successive exclusion of variable (SEV) technique that considers the significance of the variable to explain the variance through student-test. Thus the variables with minimum ‘t’ value were excluded during the regression [66]. Further, to be valid under REACH, the model was evaluated through external validation. All the models were derived and validated using Microsoft Excel 07 and MINITAB software.

**Result and Discussion**

The Overton-Meyer relationship or simply Overton rule, as stated by Meyer and Hemmi [39], is the relationship between any biological activity (SP) like tadpole narcosis in the present study and the partition coefficient log P and is as follows:

\[
\log SP = a \times \log P + C
\]  

(5)

Abraham and Rafols’s (1995) model based on log P for the set of 123 compounds is found as

\[
\log \left(1/C_{nar}\right) = 1.272 + 0.780 \pm 0.035 \log P_{oct}
\]  

(6)

\[n = 123, r = 0.9083, SD = 0.504, F = 282.7\]

When Vx was incorporated in the above regression model the correlation coefficient was increased significantly (Eq. 7). The parameter V_{x} is a solute volume parameter derived from reversed phase liquid chromatography [67]. In this regression model both the independent variables (log P_{oct} and V_{x}) are solvation oriented parameters and empirical.

\[
\log \left(1/C_{nar}\right) = 0.827 + 0.677 \pm 0.041 \log P_{oct} + 0.602 \pm 0.147 \log(1/C_{nar})
\]  

(7)

\[n = 123, r = 0.9083, SD = 0.475, F = 282.7\]

The substitution of Vx by L2, a fully theoretical parameter, improved the regression model significantly.

\[
\log \left(1/C_{nar}\right) = 1.122 + 0.736 \pm 0.033 \log P_{oct} + 0.044 \pm 0.009 \log(1/C_{nar})
\]  

(8)

\[n = 123, r = 0.9124, SD = 0.464, F = 298.3\]

Abraham and Rafols (1995) [35] used four molecular descriptors to explain the narcotic values of these 123 compounds and found nine compounds as the outliers in the plot of observed and predicted values. The outliers are triacetin, acetamide, methylurethane, nicotine, 2-propylpiperidine, urea, hexanol-1, decanol, acetal and they assigned solubility factors for the deviation of these molecules. Excluding these outliers, they obtained a regression model

\[
\log \left(1/C_{nar}\right) = 0.595 + 0.805 \pm 0.107 R_{2} - 0.725 \pm 0.135 \pi_{2}^{H} - 2.489 \pm 0.166 \sum \beta_{2}^{o} + 3.341 \pm 0.115 V_{x}
\]  

(9)

\[n = 114, r = 0.9520, SD = 0.341, F = 263.0\]

In a subsequent work Agrawal, et al. [54] added two more independent variables out of five selected topological indices in the regression model for the same data set and increased the correlation coefficient values to 0.9592.

To obtain an optimized regression model by using Abraham and Rafols’s parameters with topological parameters reported by Agrawal, et al. and L2, the data of all 114 compounds were subjected to multiple regression analysis. The Abraham and Rafols’s parameters are derived from solute-solvent inter-

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obtained (Figure 2). Considering a deviation of > 0.8 from the linearity nine outliers were identified, out of which seven are identified by Abraham and Rafols. When the data of these outliers were excluded, the regression model with 114 compounds was found to be the best among all the regression models proposed earlier (n = 114, SD = 0.3165, r = 0.96012, F = 254.7). The statistical significance of the model was further tested in terms of $K$ parameter [68]. The parameter was normally used in SEV technique to select the optimized model in situations where simultaneous increase in F and RMS value is found and computed by dividing the F value by the corresponding RMS value for a given model. The same parameter can also be employed to compare the statistical significance of two models. Increase in $K$ value suggests the statistical improvement in the model. In the present case, an increase in $K$ value to 778.9 in Eq. 11 from that of Eq. 9 ($K = 597.1$) prescribed the regression model to be better.

**Model validation**

The main objective of a QSAR model is to predict the activity of an external compound which was not used in the development of the model. As per the principles of OECD, external validation is the only way to “determine” the true predictive power of a QSAR model. In this context, the whole data set

---

**Table 4: Optimized regression model for prediction of narcotic activity of diverse organic compounds**

| $t_{max}$ | Excluded variable | $r$  | $F$     | RMS   |
|----------|-------------------|------|---------|-------|
| -0.04    | Sz                | 0.9601 | 136.280 | 0.1057 |
| 0.34     | $\chi(=B)$       | 0.9600 | 154.601 | 0.1048 |
| -0.52    | J                 | 0.9599 | 177.854 | 0.1041 |
| -0.95    | logRB             | 0.9596 | 207.512 | 0.1040 |
| 0.56     | W                 | 0.9595 | 250.518 | 0.1034 |

*aBasic regression model including all the ten descriptors (n = 114, $r = 0.9601$, $F = 121.476$, RMS = 0.1068)*.

---

Figure 2: Plot of predicted $\log(1/C_{nar})$ vs. observed ones according to model 11.
was distributed into a training set and a test set. The training set was used in the development of the model and the predictive ability of the model developed was evaluated with the external test set. The distribution was done in the ratio 1:4. For this, the compounds in the whole data set were sorted according to their narcotic activity (log (1/C_nar)) and then, every fifth compound was coined for the validation set and the rest were saved for the training set. The predictive ability of the model developed was evaluated in terms of the statistical parameter $Q_{ext}^2$ and is defined as: $Q_{ext}^2 = 1 - \text{PRESS}/\text{SD}$, where PRESS is the sum of squared differences between the observed and the predicted activity for each molecule in the validation set, and SD is the sum of squared deviations between the observed activity for each molecule in the validation set and the mean observed activity of the training set [69]. After evaluation the final model with its statistical parameters is as follows:

$$\text{Log } (1/C_{\text{nar}}) = 0.317 + 1.0660 (\pm 0.1339) R^2 - 0.5453 (\pm 0.1282) \sum \pi^H \varphi - 2.5021 (\pm 0.1543) \sum \beta^0 \gamma + 3.5593 (\pm 0.1248) \sum V_x - 0.0441 (\pm 0.01252) L^2$$ (11)

Principal component analysis

To reduce the number of descriptors in QSPR/QSAR studies, principal component analysis (PCA) plays a significant role. This technique also helps in classification of the descriptors from their relationships with the derived principal components. When all the twelve descriptors have been subjected to PCA, the first PC found to explain 60.5% variance of the toxicity and cumulatively first eight PCs can explain 99% of the variance. The PCs have orthogonal relationship with each other and hence considered as good candidates for correlation analysis, albeit the resultant coefficients do not contribute much to the physical significance of the regression model. The molecular descriptors were correlated with the PCs and it is found that beyond PC3 the correlations are found to be poor (i.e. < 0.5). PC2 correlate well with $R^2$, $J$, while PC3 has a good correlationship with $2H\alpha \sum$. These parameters have generic peculiarity and hence differ from other descriptors. Most of the rest descriptors correlate well with PC1. The descriptors $L_0$, $S_z$, $W_z$, $\chi (=B)$ and log (RB) have ‘r’ values > 0.8 and $R_p \sum \pi^H \varphi$, $\sum \beta^0 \gamma$, $V_x$ and $L_2$ have 0.6 - 0.8. Though, there is no specific reasoning of this division of parameters into different groups, the generic characteristics like connectivity, solvation may have some contribution to this classification. The newly generated weight line graph indices are found to have a common characteristic domain with the connectivity parameters.

With an aim to utilize the PCs for the prediction of toxicity, seven PCs were subjected to regression analysis with the parameters $Q_{ext}^2$ and $R^2$ for the training set are close to each other. This is further supported by the evenly distributed data points on both sides of the dashed line in Figure 3.
narcosis activity. Using SEV technique as reported earlier, the optimized equation was obtained with an r = 0.9382 and F = 158.781, which are lower than the model proposed earlier (r = 0.96012, F = 254.7).

Conclusion

Albeit narcotic behavior is a complex phenomenon in aquatic animals, the structure of the narcotic plays an important role. The parameters derived from solute-solvent interactions are the most important factor in QSAR as these parameters explain the transportability of the compounds in the body fluid and specific interactions with the biomolecules. However, the role of topological parameters, which can be obtained directly from molecular structure, cannot be ruled out. Parameters derived from line graphs, which are already reported to explain many physical characteristics of carboneous compounds, the weighted line graphs are found to be excellent additives with empirical parameters in QSAR.

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