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Solid-state polymer adsorption for surface modification: The role of molecular weight

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Hypothesis: Solid-state polymer adsorption offers a distinct approach for surface modification. These ultrathin, so-called Guiselin layers can easily be obtained by placing a polymer melt in contact with an interface, followed by a removal of the non-adsorbed layer with a good solvent. While the mechanism of formation has been well established for Guiselin layers, their stability, crucial from the perspective of materials applications, is not. The stability is a trade-off in the entropic penalty between cooperative detachment of the number of segments directly adsorbed on the substrate and consecutively pinned monomers.

Experiments: Experimental model systems of Guiselin layers of polystyrene (PS) on silicon wafers with native oxide layer on top were employed. The stability of the adsorbed layers was studied as a function of PS molecular weight and polydispersibility by various microscopic and spectroscopic tools as well as quasi-static contact angle measurements.

Findings: Adsorbed layers from low molecular weight PS were disrupted with typical spinodal decomposition patterns whereas high molecular weight (>500 kDa) PS resulted in stable, continuous layers. Moreover, we show that Guiselin layers offer an enticing way to modify a surface, as demonstrated by adsorbed PS that imparts a hydrophobic character to initially hydrophilic silicon wafers.

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1. Introduction

Supported ultrathin polymer films are integral in versatile applications such as sensors, biomedical devices, organic solar cells, and organic thin film transistors [1–5]. Films with minute thickness can also be viewed as a means to modify the chemical and physical properties of a surface [6]. Film deposition can be performed by several methods, e.g., spin-coating, Langmuir-Blodgett deposition, chemical grafting approaches, [7–9] and combining self-assembled monolayers with binding a polymer layer [10–15].

Polymer adsorption is an attractive, yet little-investigated tool for actual film deposition or practical surface modification, as the vast literature on adsorption focuses largely on fundamental phenomena [16–22]. It is based on thermodynamic strive of polymer enrichment at interfaces, thereby omitting complex chemical reaction pathways, as shown in the pioneering works of McCarthy and coworkers who suggested to exploit adsorption as an actual means for surface modification for standard substrates like silica [23–28].

Recently, our group has established generic approaches for heterogeneous surface modification of lignocellulosic surfaces by polymer adsorption from aqueous and aprotic solvents [10,29,30]. The reason for the relatively esoteric status of utilizing adsorption as a modification medium lies perhaps in the outcome: polymer layers adsorbed from a solvent often exhibit incomplete coverage and their attachment on the substrate can be unsatisfactory in applications where long-term stability is required [29]. To tackle these deficiencies, we propose the utilization of lesser known solid-state polymer adsorption in the form of so-called Guiselin layers where the coverage is usually more comprehensive and the attachment firmer [31–33].

The formation of Guiselin layers has been well established since their existence was first proposed in 1992 [31]. A bulky polymer thin film larger than a few radii of gyration \( R_g \) is placed in contact with a substrate and annealed above the glass transition temperature \( T_g \) or melting point of the polymer, resulting in an irreversibly adsorbed ultrathin layer that remains on the substrate interface after the excess polymer has been rinsed out by successive washing with a good solvent [16,17,22]. Guiselin layers have nearly exclusively been investigated within the realm of fundamental polymer physics, e.g., to unveil the adsorption mechanism or the formalism of the adsorption kinetics [17–19]. So far, the effect of the adsorbed layers on a number of materials properties has been investigated: glass transition, [22,34,35] diffusion coefficients, [36] thermal expansivity, [37] viscosity, [38] and crystallization ratio [39,40]. In this study, we want to transfer the concept of Guiselin layers into the territory of surface modification by exploring the film stability aside the standard properties, such as film thickness, coverage, and contact angle. Indeed, a crucial, yet not a fully addressed issue of Guiselin layers is their stability as ultrathin polymer films are susceptible to rupture during various treatments, e.g., elevated temperatures and solvent exposure which are inevitably present during the layer formation. This is particularly important given the issues with reproducibility of Guiselin layers, recently pointed out by Thees et al. [32]. In addition, Gin et al. [17] have observed a density difference within the adsorbed polymer layer on a silicon substrate, prompting an inner area of higher density with a more flattened conformation named a ‘lone flattened layer’, [19] and an outer bulklike lower density area named a ‘loosely adsorbed layer’. These flattened chains inhibit the penetration of free molecules causing autohydrophobic dewetting of thin polymer films at free polymer-adsorbed polymer interface, as shown in the comprehensive studies by Jiang et al. [41,42] Beena Unni et al. [43] in turn, showed that the adsorbed film stability significantly depends on the solvent used for washing out the excess polymer.

As a model system in this study, we use Guiselin layers of polystyrene (PS) on standard Si/SiO\(_x\) substrates. The idea is to show how Guiselin layers can be utilized as a means of stable surface hydrophobization of the silicon wafers. Concerning the polymer properties, we focused on molecular weight (\( M_w \)) and its effects on the formation and stability of PS Guiselin layers on silica substrates. The study differs from the existing stability studies on Guiselin layers [41–43] by its pragmatic approach as we pay particular attention to the relationship between film stability and the newly introduced hydrophobic properties.

2. Experimental parts

2.1. Materials

Toluene (>99.3%) and four PS grades were purchased from Sigma-Aldrich and used without further purification. The monodisperse PS grades are coded as PS30k (\( M_w \sim 32 000\), \( M_n \sim 31 000 \)) and PS560k (\( M_w \sim 524 000\), \( M_n \sim 502 000 \)). The polydispersity PS grades that are coded as PS53k (average \( M_w \sim 35 000 \)) and PS192k(\( M_w \sim 32 000\), PDI – 3.4) were analyzed by gel permeation chromatography (GPC), clearly showing bimodal distribution for PS35k (Table 1 and Figure S1 in Supplementary Information). Silylated wafers with a native oxide layer \( 1.7 \pm 0.2 \) nm, as measured by spectroscopic ellipsometry, were purchased from Okmetic Oy (Finland). Water used in this study was Milli-Q water (resistivity of 18.2 M\( \Omega \) cm, at 25 °C). All chemicals were used without further purification.

2.2. Preparation of substrate surface

The silicon wafers were cut to approximately \( 10 \times 10 \) mm prior to use. The wafers were cleansed by successive ultrasonic cleaning in Milli-Q water, acetone, isopropanol, and Milli-Q water. Subsequently, the wafers were treated with mild N\(_2\) purge and further cleaned in UV/ozone chamber (Bioforce Nanosciences Inc., California, USA) for 15 min. The UV/ozone cleaning procedure has been shown to be an effective method to rapidly remove a variety of contaminants from surfaces and effectively decompose hydrocarbons. [44]

2.3. Deposition of polystyrene ultrathin films

The cleaned wafers were again purged with N\(_2\) stream for dust removal and spin-coated with fresh toluene for final cleaning prior to film deposition. The atactic PS solutions were prepared in toluene (20 g/L). Thin films (thickness \( > \) few \( R_g \)) were deposited by spin-coating (WS-650SX-6NPP/LITE, Laurell Technologies) the solutions at 4000 rpm for 90 s. The spin-coated films were annealed at 150 °C for 24 h under vacuum to promote adsorption and ensure equilibrium. [22, 38] Subsequently, the films were transferred to a desiccator under air for cooling to ambient temperature. Further, the toluene leaching of the spin-coated polymer films was carried out in a systematic way by immersing the films every 10 min in 20 mL with fresh toluene at room temperature. The leaching procedure was successively carried out in one-hour total, i.e., for 6 consecutive 10 min spans. Afterwards, the films were dried in vacuum oven at room temperature to remove solvent residue before analyses.

2.4. X-ray photoelectron spectroscopy (XPS)

XPS was performed in AXIS Ultra instrument (Kratos Analytical, UK). The samples were mounted on a linear sample holder with UHV compatible carbon tape and pre-evacuated overnight. A fresh
CompleteEASE (ver. 6.51) software. For the fitting of the real part measurements were performed in the spectral range from 245 to 1690 nm wavelength. The data evaluation was carried out using a Cauchy model was assumed:

\[ n(\lambda) = A + \frac{B}{\lambda^2} + \frac{C}{\lambda^4} \]  

where \( \lambda \) is the wavelength of radiation in micrometers. A, B and C are the Cauchy coefficients and they are all fitted as positive values. The imaginary part for the complex refractive index was numerically fit to the Cauchy model using 9 points.

Table 1

| Sample code | Peaks | \( M_w/Da \) | \( M_n/Da \) | PDI |
|-------------|-------|--------------|--------------|-----|
| PS35k       | peak 1| 1203         | 815          | 1.48|
|             | peak 2| 106 550      | 53 920       | 1.98|
| PS192k      | peak 2| 173 077      | 50 939       | 3.40|

The surface topography of the spin-coated and residual films was collected through a Multiscope 8 AFM from Bruker AXS Inc. (Madison, WI, USA) in air. Images were taken with a J scanner in tapping mode using NSC15/AIBS silicon cantilevers from Mikromasch (radius of 8 nm, resonance frequency 325 kHz, Tallinn, Estonia). A minimum of three images were taken per sample, and scans were performed over several portions of the films. Other than a simple first order flattening, no image processing was carried out. For image analysis, the extracted images were subjected to ImageJ thresholding by Otsu's method producing binary images, in which the adsorbed layers (crests) and void spaces (troughs) correspond to white and black areas, respectively.

### 2.6. Atomic force microscopy (AFM)

The experiments were performed using a scattering-type scanning near-field optical microscope (s-SNOM) from Neaspec GmbH (Germany). In the setup, the atomic force microscope (AFM) works in tapping mode providing topography images of the sample surfaces. The AFM tips (Arrow NC-Pt from Nanoworld, Germany) were made of a silicon core that was coated with a Pt:Ir alloy (radius ~ 20 nm, resonance frequency 285 kHz). The tapping amplitude was maintained at 30 nm when the tip was in contact with the sample surface. The metal alloy coating of the AFM tip makes it suitable for nano-FTIR spectroscopy and infrared pseudo-heterodyne imaging. A description of these methods can be found in these references. Importantly, the pseudo-heterodyne detection technique allows a background-free accumulation of near-field optical images and provides images of both the “amplitude” relating to the reflectivity and the “phase” relati1ng to the infrared absorption. These optical images are acquired simultaneously while obtaining the AFM topography image.

With s-SNOM the complex scattering coefficient can be obtained by a Fourier transform of the interferogram of the signal for each harmonic of the vibration of the AFM tip, and the second harmonic was used in this study. To obtain s-SNOM images a scanning speed of 12.3 ms per pixel was employed and the AFM tip was illuminated by a quantum cascade laser (QCL) at 1500 cm\(^{-1}\) set to 0.3 mW output power.

The nano-FTIR spectra were acquired with a tuneable femtosecond broadband laser (at repetition frequency 80 MHz, output power 0.14 mW, spectral range 950–1950 cm\(^{-1}\)), generating the broadband IR light by employing DFG crystals. The nano-FTIR spec-
tra have a spectral resolution of 12 cm$^{-1}$, the number of scans was 10 with 9.8 ms integration time per scan. To compensate for the wavenumber-dependent laser energy, water vapor, carbon dioxide, absorption in optical components in the system etc., the nano-FTIR spectra were normalized to a background spectrum acquired at a silicon wafer using a reference calibration grating TGQ1 sample. All AFM and s-SNOM images (including all modes) were levelled and plane-fitted using the Gwyddion software (v. 2.55 for Windows - Czech Metrology Institute, Czech Republic).[52]

2.8. Contact angle goniometry

Contact angles of the modified surfaces with water (static, advancing and receding) were measured using a Theta Flex optical tensiometer (Biolin Scientific, Sweden). Static water contact angle was recorded at 10 s after placing a sensile water drop on the sample surface. Quasi-static contact angles, i.e., advancing and receding contact angle, were measured using needle-in-the-sessile-drop method.[53] Contact angles were measured with two points per substrate. At least triplicates were done for each batch of samples. The reported average values and relative standard deviations were determined based on the abovementioned six points.

3. Results and discussion

3.1. Surface chemistry and morphology

Fig. 1a represents the XPS wide-range spectra with conspicuous carbon emission indicating the irreversibly adsorbed PS layer on Si/SiO$_x$ substrate after annealing at 150 °C and subsequent solvent leaching. High-resolution XPS data in Fig. 1b depicts the adsorbed PS on the silica surface. It is noteworthy, that the band integration for 560 K was clearly higher than for other samples. In our analysis, all the spectra were referenced to the most intense PS peak present at 284.8 eV where the prominent contribution of saturated C 1s emission is typically identifiable in hydrocarbons.[16,54,55] The shake-up peak, corresponding to the π→π* transition of the aromatic ring in PS with a binding energy between 291.0 and 293.0 eV, is also clearly discernible especially for the layer of PS560k. The quantitative analysis performed on the spectra of the C 1s band is suggesting increased adsorption onto silicon surface with different PS molecular weights. However, the PS192k intensity is slightly lower than PS35k, which is likely associated with the specific surface coverage in the XPS measuring area. Nevertheless, silicon emission (Si 2p at 100 eV) is distinctive for all samples after adsorption in the wide-scan spectra, which indicates that the adsorbed Guiselin layers are not thicker than few nanometers because the escape depth of photoelectrons in XPS is at most 3–10 nm. The minute thickness was more precisely demonstrated by ellipsometry (see, Table 2). It is noteworthy that after spin-coated and annealed PS on silica – that is, before annealing to form the Guiselin layers and subsequent rinsing – the silicon emission is not observed in the XPS spectra (Figure S2). The reason for the lack of the silicon contribution is that the spin-coated PS layers are thicker than the probing depth of the XPS, i.e., >10 nm (see, Table 2).

It is also notable that the background of Si 2p and Si 2s emission in the region between 100 and 200 eV changes with different $M_w$ of PS after adsorption (Fig. 1a inset). According to the formalism of peak shape on surface morphology, as described by Tougaard et al.[56] the shape of the band background is reflecting the extent of PS coverage on silica surface. In Fig. 1a, the adsorbed PS layer with PS30k and PS35k represented similar results with raised silicon band background, suggesting a partial surface coverage by the adsorbed PS. Meanwhile, the silicon contribution in PS560k is also visible, but the signal is dampened by the layer of PS covering the silicon surface. Moreover, the loss of the O K LL Auger signal at around 1000 eV suggests a fully covered substrate, as the oxygen Auger electron escape depth is between 1 and 2 nm. [57] The emitted electrons are prevented from escaping by a continuous layer of PS560k compared to the PS30k and PS35k samples where the Auger signal does appear. In the same vein, the presence of Auger signal of the adsorbed PS192k is indicative of a partial coverage as the Auger electrons can still escape.

AFM topographies of the samples having the adsorbed PS after annealing and subsequent solvent rinsing with varying $M_w$ are consistently supporting the XPS observations of either partial or full coverage. Morphologically, a typical spinodal dewetting scenario,[58–60] i.e., the breakup of a thin film occurring through the growth of uniformly distributed surface undulations, was observed when $M_w$ values below 192 kDa were used in the process (Fig. 2 a–c). Generally, the film stability increases as the polymer $M_w$ increases, that is, the observation of crests and troughs is less dominant. In contrast to low $M_w$ PS, the adsorbed layer of PS560k (Fig. 2d) is a smooth film with topographic roughness at around 3 Å (Table 2). Our findings of low $M_w$ PS are at odds with several other reports on homogeneous layers after solid-state

![Figure 1](image-url)

Fig. 1. X-ray photoelectron spectroscopic wide scan of adsorbed PS thin films with different $M_w$ after solvent leaching after annealing at 150 °C for 24 h, as well as the insets presenting the silicon peak background, and O K LL and O 2s peaks (a), and high-resolution C 1s scan of the adsorbed PS thin films (b).
adsorption where a full coverage with a certain thickness, as measured by ellipsometry, is at least implicitly described [16,17,61,62]. As reported previously, the formation of Guiselin layers is governed by a monomer pinning mechanism (molecular motion), interfacial potential, available adsorption sites, and annealing time [32,33,61]. As the extent of adsorption increases, the space available for monomer pinning reduces. The chains adsorbed at low surface coverage create a potential opposing the growth of the interfacial layer. New chains need to stretch before diffusing through the layer formed by the molecules at surface. In this regard, the correlated reduction in the number of allowed configurations yields a severe entropy loss [63]. From the point of view of free energy, the formation of the adsorbed layer is determined by the competition between a gain in adsorption energy (monomer pinning) and the loss of conformational entropy of the chain [61,63]. Collectively, a homogeneous and flat polymer thin layer with a designated thickness after adsorption can be manipulated within the Guiselin layer construction, as confirmed by PS560k (Fig. 2 (d)). However, the conformational entropy of low molecular weight polymer in melt is greater than high-molecular-weight polymer, resulting in higher instability of the adsorbed film of PS30k, PS35k, and PS192k (Fig. 2 (a-c)). Higher instability is also evident from the ellipsometry data, showing apparently thinner layers for lower Mₗ PS grades (see, Table 2), in line with the AFM images of Fig. 2.

A bicontinuous, partially dewetted surface of the adsorbed layers on Si/SiOₓ substrate was also observed by Napolitano et al. [18,64] In their studies, a short annealing time (no exact annealing time mentioned) was associated to the rupture surface. However, an annealing time of 24 h in our study followed the criterion on equilibrium, as postulated for polymer chain conformations on a solid substrate using polymers of diverse Mₗ (123-2000 kDa) [17,22]. Periodically distributed crests and troughs were found in the adsorbed layer of monodisperse PS30k (Fig. 2(a)) after annealing and toluene solvent leaching. This is supported by Jiang et al. [41] observing the occurrence of spinodal dewetting of the adsorbed layer when the used polymer Mₗ is below critical Mₗ = 123 kDa. Jiang et al. postulated that the loosely adsorbed chains with short chains (Mₗ < 50 kDa) could be easily removed during solvent leaching due to their low desorption energy upon the small number of segment-surface contacts. It is also apparent that the single crest area was bigger with polydisperse PS35k (Fig. 2(b)). This could be induced by the adsorption of long molecules (larger Mₗ) as a ‘connector’ suppressed dewetting in the case of PS35k, as reported by Raphaël and de Gennes, [65] Reiter et al., [66,67] as well as Koga and coworkers [41,42]. However, the bimodal Mₗ distribution of PS35k with both Mₗ are below 123 kDa, which could not aid to form a full surface coverage after solvent rinsing.

Hence, it is surprising to observe a dimple structure when a polydisperse PS with Mₗ of 192 kDa is used, though the spinodal dewetting behavior was also observed when a monodisperse PS with Mₗ of 136 kDa (higher than 123 kDa in Jiang’s study) was used by Beena Unni et al. [43] Nevertheless, it is apparent that the continuous phase of the adsorbed layer increased which is revealed by an increased total surface coverage to around 80%. We may conjecture that Mₗ and polydispersity both play a significant role in surface dewetting after polymer adsorption. Given the trough depth of around 5 nm in Fig. 2(c), we scrutinized the possible existence of highly dense adsorption layer in the trough area in the following section.

3.2. Nanoscale infrared imaging and spectroscopy

Nano-FTIR, a recently developed technique, with a nanoscale-level spatial resolution by combining IR spectroscopy and s-SNOM was employed to study the surface coverage at 20 nm length scale [68]. To obtain a reference for a nano-FTIR spectrum of PS, the adsorbed layer of PS560k was studied since it formed a
continuous and a relatively thick layer on the substrate in comparison to the other samples. The obtained nano-FTIR spectra (Fig. 3) displayed two characteristic IR absorbance bands of PS at around 1450 and 1500 cm\(^{-1}\), demonstrating a good agreement between the band positions in nano-FTIR and conventional IR spectroscopies [69–71]. Despite the extremely low thickness of the PS film (ca. 5 nm), clear bands were observed, hence highlighting the capability of the instrument to provide spectra of ultrathin films with a spatial resolution on the nanoscale. Nevertheless, due to the relative minute thickness of the PS film, the signal-to-noise ratio in the acquired nano-FTIR spectra is sufficiently low to unambiguously observe other PS bands in this spectral region.

Since s-SNOM imaging offers an excellent optical spatial resolution of around 20 nm, it was possible to study the distribution of PS192k over the Si wafer. Based on the PS absorption peak in nano-FTIR spectrum of the PS560k sample above (Fig. 3), the band at around 1450 cm\(^{-1}\) in the spectrum is more intense in contrast to the band at 1500 cm\(^{-1}\), whereas the laser energy at 1450 cm\(^{-1}\) is significantly lower. Therefore, s-SNOM imaging was performed at 1500 cm\(^{-1}\) wavenumbers to investigate the possible optical (and thus chemical) contrast representing surface coverage.

The acquired s-SNOM optical images reveal a clear heterogeneity in the amplitude (Fig. 4(c)) as well as in the phase image (Fig. 4(d)) between the Si wafer and polystyrene. The Si wafer is a highly reflective surface and hence the polystyrene domains in PS192k (represented by higher areas in the AFM topography image) exhibit a lower reflectivity in the amplitude image, as revealed by lower amplitude values. In contrast, these polystyrene domains have a higher phase, which is indicative of an enhanced IR absorption at 1500 cm\(^{-1}\). The acquired images in Fig. 4 (AFM and s-SNOM) correspond very well to each other, thus confirming that polystyrene in PS192k is concentrated in domains of ca. 3–5 nm height (based on AFM data) separated with areas depleted of polystyrene. However, because the s-SNOM images show only relative changes, there is a possibility that trace amounts of polystyrene remain on the bottom of the troughs. In principle, nano-FTIR spectra could reveal the presence of polystyrene in the troughs, but for the PS192k film that is thinner than the PS560k film in Fig. 3, no signal above the noise level was observed neither at the domains nor in the troughs. In general, the chemical contrast seen in the obtained s-SNOM images agrees with the above XPS data and indicate that low \(M_n\) polystyrene, contrary to PS560k, has tendency to disrupt the continuous layer, following the spin-coating on Si wafers and the subsequent heat and solvent treatments, as discussed previously.

3.3. Layer thickness

The thickness of the layers before and after adsorption was measured with spectroscopic ellipsometry considering a simple multilayer model, air/PS/SiO\(_x\)/Si (substrate). More detailed parameters and fittings regarding the model are revealed in Figure S3. The accuracy of the thickness evaluation by ellipsometry was complemented by AFM scratching approach in the case of smooth, continuous films from PS560k. According to AFM, the thickness of the adsorbed PS560k was ca. 5 nm (Figure S4), which is comparable to the value from ellipsometry, i.e., 5.4 nm. However, the high roughness value (Table 2) of the rupture surface with lower \(M_n\) might cause less accuracy of the fitted thickness data [18]. With regard to the rupture surfaces, the thickness is closer to the average thickness rather than the maximum thickness due to the existence of troughs or voids on the surface. Moreover, the sensitivity of ellipsometry is generally considered to decrease in films below 10 nm. Nevertheless, ellipsometry was able to distinguish thickness values ranging from 1 nm to 5 nm (Table 2).

3.4. Film stability

The film stability is increasingly important when considering the possible applications of the adsorbed layer. Heterogeneous dewetting may cause film instability at all stages of Guislain layer formation, i.e., spin-coating, annealing, [34] and solvent leaching [43,72]. It is triggered by the presence of omnipresent heterogeneities, e.g., dust particles or defects at the substrate surface [73]. Thus, in this study, the silicon wafers were carefully cleansed to remove hydrocarbon contaminants and dust particles on the surface to minimize the possibility of heterogeneous dewetting. Without major heterogeneities, the films may dewet in a spinodal scenario, [58–60] i.e., by the amplification of capillary wave fluctuations of specific wavelength, often enhanced by the native oxide (SiO\(_x\)) layer at the substrate-polymer interface. As mentioned earlier, annealing time with temperature above the \(T_g\) was agreed as one of the crucial factors to facilitate the film stability on the substrate. Apart from that, solvent washing including washing time and solvent choice is debated as another factor provoking the morphology of the adsorbed layer. For example, Beena Unni et al. [43] found that the adsorbed layers underwent spinodal dewetting at different time scales of rinsing upon solvent polarity. The use of different solvents results in varied transition layer thickness below which the flat film turns to spinodal dewetting. When toluene was used as a good solvent for PS, the experimental transition thickness was determined as ca. 2.8 nm. Conversely, Davis et al. [72] have recently demonstrated that the thickness of the adsorbed PS layer is insensitive to the used solvent type.

All in all, the film stability can be originated and mathematically described by the Lifshitz-van der Waals interaction potential. The interfacial potential is reconstructed from references [59,73] in Fig. 5 for SiO\(_x\)/PS/air system with a 1.7 nm SiO\(_x\) layer, i.e., similar to the thickness of a native oxide layer in silicon wafers used in this study. As established in the system of SiO\(_x\)/PS/air, the interfacial potential can be written: [59,74]

\[
\phi(h) = \left(\frac{A_{0} \pi r \sigma}{12h(h + d_{SiOx})}\right) + \left(\frac{A_{0} \pi r \sigma}{12h}\right) - \left(\frac{A_{0} \pi r \sigma}{12h + d_{SiOx}}\right)
\]

where, \(h\) is the film thickness; \(d_{SiOx}\) is the thickness of silicon oxide layer, and \(A_{0}(PS/\text{Air})\) and \(A_{0}(SiOx/\text{Air})\) stand for the effective
Hamaker constants of the Si/PS/Air and the SiOx/PS/Air systems, respectively. According to literature, the $A_{SiOx/PS/Air}$ is not well defined, ranging from $-0.22$ to $1.6 \times 10^{-20}$ J [62]. In this case, the curvature (second derivative) of the interfacial potential is normally considered, defining that at $\Phi''$ $(h) > 0$ are metastable films and do not dewet whereas at $\Phi''$ $(h) < 0$ are unstable [75]. In this study, the thickness of SiO$_x$ layer at ca. 1.7 nm does not cause dewetting when the PS film thickness is over 3 nm (film thickness after spin-coating in Table 2) due to the sign of $\Phi''$ $(h)$ which is positive for a metastable film (Fig. 5) [62]. As another non-negligible postulation, the residual stress generated after spin-coating is also a possible driving force for dewetting. Reiter et al. [76] found that residual stresses are estimated to be on the same order of magnitude as the acting capillary forces on the course of annealing above $T_c$. However, the heterogeneities, the presence of the native silica layer, and the residual stress from spin-coating would only cause dewetting throughout the entire spin-coated film, instead of merely occurring in the underlying layer in the vicinity of the silica surface (i.e., in the Guiselin layer). Here, the spin-coated films after annealing were intact according to AFM topography (Figure S5).

In contrast to the whole spin-coated film, the thinner Guiselin layer may be more labile, particularly upon its formation under elevated temperature. The adsorbed layer may play a distinct role of a mobile phase apart from the bulk material causing density variation [17,19] and surface undulations during annealing [21,37]. As a result, a property difference abounds between the adsorbing/adsorbed layer and the outer bulklike part. In other words, we must consider a system SiOx/PS (Guiselin)/PS (bulk)/air instead of the conventional SiOx/PS/Air (eq. 2) during solid state adsorption. Recently, Li et al. [77,78] showed that the chain evolution of the irreversibly adsorbed layer during annealing process caused entropic inequivalence, resulting in macroscopic dewetting of the entire thin film (thickness of 250 nm) upon prolonged annealing of 120 h. The relatively short annealing time of 24 h in this study may merely promote the dewetting of the adsorbed Guiselin layer, not the dewetting of the whole spin-coated film. According to literature, [16,18] the PS30k and PS35k will show thickness equilibrium (i.e., namely the asymptotic plateau value of the residue thickness upon long annealing times) of 2.6–2.8 nm, which is comparable to the ellipsometry study in our case. Supported by the interfacial energy, the adsorbed PS30k and PS35k with thickness below 3 nm (Table 2) presented a dewetting pattern at the presence of SiO$_x$ layer at ca. 1.7 nm, as revealed in Fig. 2(a) and 2(b), respectively. Simultaneously, it was postulated that the adsorbed Guiselin layer could partially ‘migrate’ to bulk phase during film cooling stage after annealing due to its greater entropy gain for lower molecular weight polymers [79]. As judged by the adsorbed layer thickness on the interfacial energy theory, the adsorbed PS192k (Fig. 2(c)) should be as stable as PS560k (Fig. 2(d)). However, the low molecular fraction and polydispersity of PS192k (Figure S1) may interfere with the adsorption behavior. More systematic study on the film stability of using polydisperse polymers should be considered. Furthermore, the stability of PS560k Guiselin layer after the deposition was verified to check the effect of annealing on the stability of the adsorbed layers. The PS560k Guiselin layer was post-annealed for another 24 h and found unstable with heterogeneous dewetting (Figure S6) instead of wavelength-dependent spinodal dewetting. As the heterogeneous dewetting resulted in a relatively low abundance of troughs, however, its effect on the contact angle with water (Table 3) – i.e., on the actual extent of surface modification – was minimal. All this suggests that the adsorbed Guiselin layer is a stable and ‘not dead’ adsorbed layer on the substrate, which is consistent with the dewetting or instability of the formed Guiselin layer after post-annealing [62]. Therefore, the adsorbed Guiselin layer is a ‘living layer’ under the annealing process, which may easily promote the occurrence of dewetting in the adsorbed polymer layer especially with low molecular weight and polydispersity.

### 3.5. Water contact angles

The hydrophobizing effect of the adsorbed polymer layer is prominent as the water contact angles of the treated silicon wafers show in Table 3 and Figure S7. The static contact angles of the adsorbed polymers with different PS $M_w$ values are around 90°. This is a prominent comparison with the cleansed hydrophilic silicon wafer which, when untreated, is effectively wetted by water (Figure S7a) and showing near-zero contact angles (below 5°) – a fact well established [80,81]. It should be mentioned that the water contact angle of the pristine, bulky PS directly after spin-coating (ca. 100 nm) is around 92° in this study (Figure S7b, Table 3), serving as a comparative reference for the adsorbed thin layers. It is clear that the Guiselin layers of PS192k and PS560k present a value close to that of a thick pristine PS film (ca. 100 nm), indicating a successful alteration of surface hydrophobicity, indeed a reversal from hydrophilic to hydrophobic. The slight variation of the measured static contact angle of PS film in air can be associated as polymer thickness-dependent effect, governed by the long-range van der Waals forces from the underlying substrate and the topography roughness [82]. This unambiguously illustrates the efficiency of Guiselin layers, particularly with high $M_w$ PS560k, to completely alter the surface properties of silicon wafers from highly hydrophilic to hydrophobic. One could think of utilizing Guiselin layers as an alternative to self-assembled monolayers.

### Table 3

| Thin films on silica | Static contact angle* $h_a$, $\circ h_r$ | Advancing / receding contact angles, $h_a/h_r$ (°) | Contact angle hysteresis, $\theta_h - \theta_a$ (°) |
|---------------------|--------------------------------------|-----------------------------------------------|---------------------------------------------|
| PS30k               | 89.0 ± 1.8                           | 94.3 ± 0.1/72.5 ± 0.3                           | 21.8                                       |
| PS35k               | 90.9 ± 0.3                           | 95.2 ± 0.2/76.2 ± 0.5                           | 19.0                                       |
| PS192k              | 91.2 ± 0.4                           | 97.2 ± 0.4/82.1 ± 0.4                           | 15.5                                       |
| PS560k              | 91.0 ± 0.4                           | 96.9 ± 0.3/82.1 ± 0.2                           | 14.8                                       |
| PS560k, spin-coating| 92.1 ± 0.2                           | 96.2 ± 0.2/82.1 ± 0.1                           | 14.1                                       |
| PS560k Guiselin layer_post-annealing| 92.1 ± 0.2| 95.8 ± 0.3/81.4 ± 0.2 | 14.4 |

* The contact angle is measured 10 s after the sessile drop was placed on the sample surface.
### Table 4
Comparison among different surface modification approach for solid substrate in the viewpoint of their processes and main features.

| Process [reference] | Description | Requirements on the substrate | Requires custom-built instrument |
|---------------------|-------------|-------------------------------|----------------------------------|
| Plasma treatment [84] | Bombarding substrate with electrons and ions under plasma | – | Yes |
| Coating technologies | Wide variety of methods to form a film by precipitation on a solid surface; usually results in thick and rough films | – | Not necessarily |
| Spin coating [85] | Sub-category of coating for controlled deposition of smooth, ultrathin films | Must be non-porous | Yes |
| Chemical vapor deposition (CVD) [86] | Wide variety of deposition techniques of gaseous substances on solid surfaces | Must be reactive with the coating species | Yes (for controlled deposition) |
| Self-assembled monolayers [83] | Chemisorption of surfactant head groups, followed by a tail organization | Must be reactive with the head groups | No |
| Langmuir–Blodgett technology [87] | Monolayer by monolayer deposition of (usually) organic material from liquid-gas interface to solid substrate | Planar substrate required | Yes |
| Layer-by-layer deposition [88] | Films formation by depositing alternating layers of oppositely charged polyelectrolytes or colloidal particles | Requires attractive interactions (often opposite charge) with coating polymers | Not necessary (labor-intensive without one) |
| Guiselin layer deposition [this study] | Thick film deposition followed by annealing above glass transition temperature and fresh solvent leaching | No | No |

(SAMs) which generally require a reactive interface to form [83]. A comparison among different modification approach for a solid surface is listed in Table 4. Since the formation of Guiselin layers is driven by thermodynamic strive, [18] the chemical match between the substrate and the coating is irrelevant, i.e., no demanding requirement of reactive substrate. As a result, virtually any substrate/polymer combination is feasible as long as the substrate can withstand the annealing and solvent leaching.

The advancing and receding contact angles were measured for the evaluation of the surface roughness and chemical heterogeneity, as revealed by hysteresis. Table 3 shows how the water contact angle hysteresis decreases as the surface coverage of PS increases with high $M_w$. As a general trend, the water contact angle hysteresis showed a decreasing tendency with higher surface coverage upon the increase of the applied PS $M_w$. This is a result of the surface rupture of low $M_w$ polymers as observed by both AFM (Fig. 2) and XPS (Fig. 1), and altogether the hysteresis is more sensitive to the distinction among Guiselin layers of different $M_w$ than the mere contact angles. However, as indicated by the calculated water contact angles (Table S1) with the hypothesis of Cassie-Baxter state [89] using the analyzed surface coverage, the current surface dewetting pattern should fit the Wenzel state. Therefore, we may doubt the accuracy of the obtained surface coverage of AFM images in particular with PS192k (Table S1). We may still conjecture that PS segments are probably underlying the troughs and cover the rupture area as shown in AFM images, despite intensive surface coverage studies of PS192k via surface sensitive techniques, i.e., XPS and Nano-FTIR pointing otherwise.

### 4. Conclusion

In this study, the solid-state adsorption was revisited according to the well-established Guiselin process [31] with the emphasis on application for surface modification. The existence of the adsorbed layers was established by monitoring the surface chemical composition with XPS. Contrarily to several other accounts, [16,61,62] however, the adsorbed layers presented typical surface spinodal dewetting with molecular weight dependence, as revealed by AFM. The thickness of the adsorbed layers was investigated on average by ellipsometry in the range of 1.5 to 5.5 nm without considering the surface roughness. On the basis of previous investigation on the film stability upon the effect of $M_w$ in the systems of monodispersed polymers determined by interfacial potential difference and density variations within the adsorbed layer, [21,41,42,62] the spinodal dewetting during Guiselin layer forma-

**CRediT authorship contribution statement**

Wenyang Xu: Conceptualization, Methodology, Validation, Formal analysis, Investigation, Visualization, Writing – original draft.
Karl Mihhels: Investigation, Validation.
Nikolay Kotov: Investigation, Formal analysis, Validation.
Sakari Lepikko: Investigation, Validation.
Robin H.A. Ras: Methodology, Validation, Writing - review & editing, Supervision, Resources.
C. Magnus Johnson: Investigation, Formal analysis, Resources, Validation.
Torbjörn Pettersson: Conceptualization, Methodology, Resources, Writing - review & editing, Supervision, Project administration, Funding

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acquisition. \textbf{Eero Kontturi:} Conceptualization, Methodology, Resources, Writing - review & editing, Supervision, Project administration, Funding acquisition.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.jcis.2021.07.062.

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