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Cadmium contamination of sediments in the water reservoirs in Silesian Upland (southern Poland)

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Abstract

Purpose Cadmium (Cd) is considered a toxic element and its concentrations are relevant to human health and the environment. Therefore, the purpose of the study was to determine the extent to which the bottom sediments of water bodies (artificial lakes and ponds) in the Silesian Upland in southern Poland are contaminated with Cd; an attempt was also made to determine the factors that condition spatial differences in the concentration of this element between individual water bodies in the region.

Materials and methods Measurements of the Cd content in bottom sediments were carried out in 35 water bodies in southern Poland in 2011 and 2012. Depending on the surface area and morphometric characteristics, from two to nine samples representative in terms of sediment thickness were collected in each water body. Cadmium concentrations were determined for 92 0.25 g aliquots using the TD-ICP method.

Results and discussion Cadmium content in all samples (0.7–580.0 mg kg⁻¹) was higher than the natural range of concentrations for this element in the Earth’s crust (0.1–0.3 mg kg⁻¹) and the geochemical background for Poland (0.5 mg kg⁻¹) and, with a few exceptions, was also higher than the preindustrial concentration (1.0 mg kg⁻¹) and the regional geochemical background (2.5 mg kg⁻¹). Adopting natural Cd concentrations in the Earth’s crust (0.1–0.3 mg kg⁻¹) as the baseline for the geoaccumulation index (Igeo), the sediments examined can be classified as extremely and heavily contaminated (and moderately contaminated in a small number of cases). The assessment of sediment quality based on Igeo, with the regional geochemical background (2.5 mg kg⁻¹) adopted as the baseline, results in non-contaminated and moderately contaminated sediments being dominant with a far smaller number of heavily and extremely contaminated ones.

Conclusions In the case of several water bodies, Cd concentrations were at record levels that have not been found anywhere else in the world. On the basis of the Igeo, sediments of varying quality were found—from virtually uncontaminated to extremely contaminated. The Igeo index as an indicator of the quality of bottom sediments is a measure that requires careful interpretation, especially when different concentration levels regarded as natural are used for determining its value.

Keywords Bottom sediments · Cadmium · Environmental pollution · Heavy metal · Lakes · Pollution levels · Silesian Upland · Toxic metals · Trace elements · Water reservoirs

1 Introduction

Cadmium (Cd) is classified as a highly toxic heavy metal (Håkanson 1980; Bojakowska and Sokolowska 1998). Its high toxicity poses a threat to the development of flora and fauna as well as to human life and health (Bennet-Chambers et al. 1999). Cadmium toxicity may be exacerbated by the presence of other toxic metals, e.g. zinc (Świderska-Bróž 1993). Cadmium is easily absorbed and accumulates in tissues, and its main sources in our diet are fish and cereal products (Olmedo et al. 2013). Long-term Cd poisoning may lead to damage to the kidneys, liver, testes and prostate. Excessive Cd exposure may also cause anaemia, high blood pressure, circulation problems, decalcification of bones and muscle atrophy (Kabata-Pendias and Pendias 1993). Owing to its toxic...
properties, Cd is considered unnecessary for the proper functioning of living organisms.

For these reasons, Cd concentrations in the natural environment are monitored closely. Studies of bottom sediments accumulated in water bodies are of special importance since the concentration of Cd in these sediments is a good indicator of water and sediment pollution and also of environmental conditions that prevail in the catchment in question (Müller 1979; Håkanson 1980). The monitoring of Cd presence in the environment is especially justified in the case of sediments that are present in water bodies, especially where these are used for various purposes, e.g. irrigation in agriculture and the filling of fish ponds, water supply for households and industry, the extraction of mineral resources and recreation.

Studies of Cd content in the natural environment are mainly required in industrialised and urbanised areas with high population density. Such areas include the Silesian Upland in the central part of southern Poland, where the long-term exploitation of mineral resources and the growth of manufacturing industry together with urbanisation processes have resulted in the transformation of the natural environment on a scale unprecedented in Poland. In areas of historical and contemporary extraction of mineral resources (e.g. coal, lead (Pb) and zinc (Zn) ores, iron (Fe) ores), where Fe and non-ferrous metal smelting as well as other manufacturing industries developed, degradation of the natural environment has been particularly severe (Fig. 1).

The rationale for conducting studies on the Cd content in the bottom sediments of water bodies in the Silesian Upland is the insufficient recognition of the scale of environmental pollution caused by this element. To date, the presence of Cd in the bottom sediments of water bodies has been discussed in the context of limnological studies that only touched upon this subject (Jankowski et al. 2002; Ciszewski et al. 2013; Rzetala et al. 2013; Rzetala et al. 2015; Rzetala 2014, 2015b) and also studies conducted within the framework of geochemical examination of soil and surface formations, water and aquatic sediments (Lis and Pasieczna 1995a, b). In those studies, the value of the regional geochemical background for Cd in aquatic sediments was determined at 2.5 mg kg\(^{-1}\) (Lis and Pasieczna 1995b). Owing to the specific geological structure of the Silesian Upland and the presence of many mineral resources, especially Zn and Pb ores in Triassic formations, it is higher than the geochemical background for Cd in Polish aquatic sediments, which is <0.5 mg kg\(^{-1}\), and also than the threshold value for the geochemical background of aquatic sediments in Poland, which was determined at 1.8 mg kg\(^{-1}\) (Lis and Pasieczna 1995a). Similar relationships have been found if the natural Cd content in Polish soils (<0.5 mg kg\(^{-1}\)) are compared to that for soils in the Silesian Upland (1.3 mg kg\(^{-1}\), while the geochemical background for Cd in surface waters in Poland as a whole and on the Silesian Upland is the same, at <3.0 \(\mu\)g dm\(^{-3}\) (Lis and Pasieczna 1995a, b).

The purpose of the study was to determine the extent to which the bottom sediments of water bodies in the Silesian Upland in southern Poland are contaminated with Cd; an attempt was also made to determine the factors that condition spatial differences in the concentration of this element between individual water bodies in the region.

## 2 Materials and methods

The emergence of several thousand (usually small) bodies of water in the Silesian Upland was related to economic development (Rzetala and Jaguś 2012). Natural bodies of water (lakes) are virtually absent in the study area; this is a consequence of, inter alia, the progressing karstification of the substrate in part of the study area, its old-glacial relief and human impact. Anthropogenic basins filled with water dominate this area so much that the Silesian Upland, together with parts of adjacent regions, is referred to as the Upper Silesian Anthropogenic Lake District, which formed in recent decades (Rzetala and Jaguś 2012).

Artificial water bodies serve as sedimentary basins where bottom sediments are formed in which pollutants accumulate. Bottom sediments can be considered a good indicator of environmental conditions and processes that take part in the watershed and in close vicinity of water bodies (Tylmann 2004; Fajer et al. 2012; Szymczyk 2012; Rzetala et al. 2013).

Measurements of Cd content in bottom sediments were carried out in 2011 and 2012 in 35 water bodies in southern Poland (Fig. 2). The water bodies selected for study are 60 to 70 years old on average and have varying origins: flooded mineral workings, reservoirs impounded by dams and embankments, water bodies in subsidence basin and water bodies with multiple origins. Flooded mineral workings prevail (Fig. 2 (1–31)), with some basins impounded by small dams (Fig. 2 (1, 2, 15, 18 and 32)) or reshaped by subsidence processes (Fig. 2 (27–31)). The remaining impoundments and levee ponds formed due to the slight increase in the water level caused by the presence of dams or levees (Fig. 2 (32–35)). In most cases, these bodies of water are small in terms of their area and retention capacities (Table 1). The surface areas of the water bodies tested range from 0.9 to 561 ha with an arithmetic mean of 49.9 ha, a median equal to 9.0 ha (first quartile 2.9 ha, third quartile 36.7 ha) and a standard deviation of 106.8 ha. The amount of water they hold ranges from 6.0 to 94,000 dam\(^{3}\); other statistical characteristics concerning the retention capacities of the water bodies in question are as follows: 4597 dam\(^{3}\) (arithmetic mean), 202 dam\(^{3}\) (median), 57.0 dam\(^{3}\) (first quartile), 797 dam\(^{3}\) (second quartile) and 16,509 dam\(^{3}\) (standard deviation). They also vary in terms of their water quality (Table 1), which is indirectly evidenced by the variability in specific electrolytic conductivity (183.0–
Fig. 1 The distribution of surface formations (a) and the geological structure of the substrate (b) on the Silesian Upland and major mineral resource deposits, their mining and processing (after Lis and Pasieczna 1995b; PGI 2006; simplified and supplemented): 1—sands, gravels, alluvial soils, peat and silts (Holocene); 2—eolian sands, locally in dunes (Quaternary); 3—loesses, sandy loess and loess-like dusts (Quaternary); 4—clays, sands and clays with gruss, solifluctue-deluvial (Pleistocene); 5—sands, gravels and alluvial loams (Pleistocene); 6—marginal clays, loams and sands (Pleistocene); 7—outwash sands and gravels (Pleistocene); 8—end moraine gravels, sands, boulders and clays, glacial tills, glacial till waste and glacial sands and gravels (Pleistocene); 9—organodetritic limestones, sulphur-bearing limestones, gravels, clays, loams, sands, locally gypsum and lignite (Neogene); 10—limestones, marls, dolomites, mudstones, sandstones, claystones, with flints and siderite insertions, gravels, conglomerates, sands, clays and fire clays (Jurassic); 11—claystones, mudstones, sandstones, limestones, dolomites, marls, oolitic limestones, gypsum, anhydrite, epigenetic ore-bearing dolomites, iron ores (Triassic); 12—conglomerates, arkosic sandstones, mudstones, claystones (Perm); 13—sandstones, conglomerates, claystones, mudstones, coal (Carboniferous); 14—waterways and water bodies; 15—watershed; 16—actively mined coal deposits; 17—zinc and lead ore deposits; 18—power plants; 19—cement factories; 20—steelworks, smelters; 21—metallurgical processing plants or smelters (zinc and lead); 22—former smelters (zinc and lead); 23—water body labels (see Fig. 2).
16,500 μS cm$^{-1}$), nitrate concentrations (1.0–42.2 mg NO$_3$ dm$^{-3}$) and phosphate concentrations (0.0–5.1 PO$_4$ $^{3-}$ dm$^{-3}$).

Bottom sediment samples were collected using the Beeker loose sediment sampler (04.20.S.A version, manufactured by...
Eijkelkamp). The use of a sampler with a transparent sediment tube with guides, a built-in hammer, pneumatic lower closing element and several other improvements made it possible to collect sediments with a virtually undisturbed structure. Samples were additionally collected using a van Veen sampler of a capacity of 1.25 or 2.50 dm$^3$. As there were very few sediments on the rocky bottom of the Koparki water body, samples were collected directly into a polystyrene container by a team of divers. Depending on the surface area and morphometric characteristics, from 2 to 9 samples representative in terms of sediment thickness were collected in each water body (92 samples in total). Samples were collected in polystyrene containers or polyethylene bags and transported to the laboratory. During the deposition of the material collected for testing in containers (bags), mixed samples were prepared that were representative of the vertical sediment profile. The vertical sediment profile is assumed to be representative of the entire layer of sediments found on the bottom, which exhibited limited thickness in most cases. These deposits tend to be unconsolidated, often have thixotropic characteristics and may also be subject to periodic inactivation in the water mass. Mixed samples were considered to best reflect the general geoeological state of sediments, which is representative of anthropogenic lakes that are often only a few decades old.

The basic preparation of the sediment samples for laboratory tests of their geochemical composition was carried out at the Faculty of Earth Sciences of the University of Silesia. After the material had been dried at a temperature of 105 °C, it was ground in a mortar. The <0.063-mm fraction was then separated using chemically inert sieves. Cadmium concentrations were determined at Activation Laboratories Ltd. (Ancaster, Canada) from 0.25-g aliquots using the total digestion ICP method. The sample aliquots were digested with a mixture of HClO$_4$, HNO$_3$, HCl and HF at 200 °C to fuming and were then diluted with aqua regia (www.actlabs.com 2016). The sample solutions were also spiked with internal standards and further diluted and introduced into a Perkin Elmer SCIEX ELAN 6000 ICP-MS. The tests were performed at a Cd detection level of 0.5 mg kg$^{-1}$ and an upper limit of quantification of 5000 mg kg$^{-1}$. The precision and accuracy of the analyses conducted using the PERKIN ELMER Model 6000 spectrometer are as follows: (a) at the lower limit of detection ±100 %; (b) at 10 times the lower limit of detection ±15–20 % and (c) at 100 times the lower limit of detection >10 %.

The geoaccumulation index ($I_{\text{geo}}$) algorithm for geochemical sediment studies proposed by Müller (1979) was used as follows:

$$I_{\text{geo}} = \log_2 \frac{C_n}{1.5B_n}$$

where

$C_n$ is concentration of the element in question in bottom sediments, $B_n$ is geochemical background for the element in question and 1.5 is the coefficient expressing natural variation in the content of the element in question in the environment.

The $I_{\text{geo}}$ has several classes of sediment quality (Müller 1979): class 0 ($I_{\text{geo}} \leq 0.0$)—practically uncontaminated; class I—uncontaminated to moderately contaminated ($0.0 < I_{\text{geo}} < 1.0$); class II—moderately contaminated ($1.0 < I_{\text{geo}} < 2.0$); class III—moderately to heavily contaminated ($2.0 < I_{\text{geo}} < 3.0$); class IV—heavily contaminated ($3.0 < I_{\text{geo}} < 4.0$); class V—heavily to extremely contaminated ($4.0 < I_{\text{geo}} < 5.0$) and class VI—extremely contaminated ($5.0 < I_{\text{geo}}$).

The $I_{\text{geo}}$ index is widely used in sediment contamination studies (e.g. Müller 1979, 1986; Förstner and Müller 1981). Its advantage is that it offers the possibility of assessing sediment contamination in various sedimentary environments. In the case of aquatic ecosystems, this index is commonly used for marine (Özkan 2012; Attia and Ghrefat 2013; Ong et al. 2013), fluvial (Bojakowska and Sokolowska 1998; Ghrefat and Yusuf 2006; Nikolaidis et al. 2010; Kabir et al. 2011; Rabee et al. 2011; Rubin et al. 2011; Varol and Sen 2012; Yahaya et al. 2012; Kalender and Uçar 2013), lake (Zeng and Wu 2009; Choinski et al. 2010; Bing et al. 2011; Hu et al. 2011; Cáceres Choque et al. 2013; Alikaj et al. 2014;...
El-Sayed et al. 2015; Gao et al. 2015; Guo et al. 2015) and reservoir sediments (Loska and Wiechułła 2003; Saleem et al. 2012; Young et al. 2012; Bąk et al. 2013; Goher et al. 2014; Tang et al. 2014; Wang et al. 2014; Palma et al. 2015) as well as in marshes and peats (Bai et al. 2012) in various climatic zones.

Modern applications of the Igeo are also varied; even for the algorithm for calculating, it takes different forms. The essence of the problems arising in relation to interpretation is the value of the geochemical background that is used in the calculation of the index for the element in question. Some researchers base their interpretations of element concentrations in sediments on a calculation formula that uses the presence of the element in question in a certain type of rock in the Earth’s crust as the geochemical background (Loska and Wiechułła 2003; Ghrefat and Yusuf 2006; Hu et al. 2011; Saleem et al. 2012; Young et al. 2012; Goher et al. 2014; El-Sayed et al. 2015). Most commonly, the geochemical background for individual elements is used (Turekian and Wedepohl 1961; Bowen 1966; Li 2000; Li and Schoonmaker 2005). Equally popular is a solution in which the geochemical background of a region (Lis and Pasieczna 1995a, 1995b; Tylmann et al. 2011; Guo et al. 2015; Rzętała 2015a; 2015b; 2015c) or study area (Tylmann et al. 2011; Tang et al. 2014) is taken into account, as well as the preindustrial content of the element in question in the environment (Håkanson 1980).

3 Results

Cadmium was present in the bottom sediments of the water bodies in amounts ranging from 0.7 to 580.0 mg kg$^{-1}$ (Fig. 3). The minimum concentration from this range concerned a sediment sample from the Pogoria III water body, and the maximum one was for a sample from the Brzeziny water body. The lowest Cd concentrations were found in the bottom sediments of three flooded mineral workings: Gliniok (1.2 mg kg$^{-1}$), Kradziejówka (1.5–2.5 mg kg$^{-1}$) and Pławniowice (1.8–1.9 mg kg$^{-1}$). The highest concentrations were found in the bottom sediments of the Brzeziny (540–580.0 mg kg$^{-1}$), Hubertus I (274.0–444.0 mg kg$^{-1}$), Gliniak (172.0–286.0 mg kg$^{-1}$) and Morawa (119.0 mg kg$^{-1}$) water bodies.

Fig. 3 Cadmium content in the bottom sediments of the water bodies examined (water body numbering as in Fig. 2)
The arithmetic mean of Cd concentrations for individual water bodies ranged from 1.2 to 560.0 mg kg\(^{-1}\) (Fig. 4). The arithmetic mean of Cd concentrations in the bottom sediments, calculated for 35 water bodies on the basis of their individual mean values, was 52.1 mg kg\(^{-1}\) with a median equal to 14.3 mg kg\(^{-1}\) (first quartile—8.1 mg kg\(^{-1}\), third quartile—30.8 mg kg\(^{-1}\)) and a standard deviation of 113.0 mg kg\(^{-1}\). The statistical characteristics calculated for minimum concentrations of Cd in the sediments of the water bodies are as follows: 0.7–540.0 mg kg\(^{-1}\) (range of variation), 12.5 mg kg\(^{-1}\) (median), 44.2 mg kg\(^{-1}\) (arithmetic mean) and 102.2 mg kg\(^{-1}\) (standard deviation). The corresponding parameters for maximum concentrations are as follows: 1.2–580.0 mg kg\(^{-1}\) (range of variation), 14.4 mg kg\(^{-1}\) (median), 60.2 mg kg\(^{-1}\) (arithmetic mean) and 125.2 mg kg\(^{-1}\) (standard deviation).

4 Discussion

4.1 Cadmium in Silesian reservoirs

Cadmium is present in the upper layer of the Earth’s crust at a concentration of 0.1 mg kg\(^{-1}\) on average (Taylor and McLennan 1995; Li 2000; Li and Schoonmaker 2005). Turekian and Wedepohl (1961) state that this element is present at slightly higher levels (0.3 mg kg\(^{-1}\)) in some types of sedimentary rock. On the other hand, Bowen (1966) states that the Cd content in sedimentary rocks ranges from 0.035 to 0.3 mg kg\(^{-1}\). Kabata-Pendias and Pendias (1979) and Taylor (1964) state that the average Cd content in the Earth’s crust ranges from 0.15 to 0.20 mg kg\(^{-1}\). Given these natural limits to Cd content, it can be concluded that the concentration of this element in all the samples of bottom sediment in water bodies in the Silesian Upland was higher than the natural
range of concentrations for this element, often by several orders of magnitude.

In all the water bodies examined, the concentration of Cd in bottom sediments is greater than the preindustrial content of this element (1.0 mg kg\(^{-1}\)) in sediments of Eurasian and American lakes as suggested by Häkanson (1980). The Cd content found in the bottom sediments is still higher than the geochemical background for Poland determined at below 0.5 mg kg\(^{-1}\) by Lis and Pasieczna (1995a). For the sediments examined, in only a few cases is the threshold geochemical background value for Cd in aqueous sediments in Poland (1.8 mg kg\(^{-1}\)) not reached (Lis and Pasieczna 1995b). For aqueous sediments in the Upper Silesia region, the geochemical background for Cd is only slightly higher at 2.5 mg kg\(^{-1}\) (Lis and Pasieczna 1995b).

This means that the concentration of Cd in the sediments of just three water bodies examined was lower than the regional geochemical background, and in the other ones, this background level was exceeded significantly.

Causes for the variation in Cd concentrations in bottom sediments between individual water bodies include differences in natural factors (these are usually related to geological structure) but primarily relate to human pressure, particularly, land use in the catchment and air pollution. Similar considerations apply to differences in the concentration of other metals, non-metals and metalloids in the region (Rżętała 2015a; 2015b) and in other parts of the world (Astrom and Nylund 2000; O’Neill et al. 2015).

The role of anthropogenic factors that condition the accumulation of Cd in sediments is well documented by the example of the water bodies most contaminated with this element (i.e. Brzeziny, Hubertus I, Gliniak, Marawa, Stawiki). The Brzeziny water body lies in the immediate vicinity of spoil tips derived from the smelting of Pb and Zn ores. It is fed by leachate water from a spoil tip and the material deposited there also finds its way to the basin of the water body. The Hubertus I, Gliniak, Marawa and Stawiki water bodies are located in an area subject to the long-term impact of a non-ferrous metal smelter. Cd contamination is also the result of the storage of the waste produced by non-ferrous metal smelting within the basins of water bodies, its use in rehabilitation work and in the construction of transport infrastructure. On the other hand, Chechło Lake is an example of a local sedimentary basin where Cd is supplied by atmospheric deposition from a nearby non-ferrous metal smelter. The accumulation of Cd in the sediments of the aforementioned water bodies is permanent to the extent that the mobility of this element is limited by the alkalinity of the environment in which it is deposited. However, the catchments of these water bodies are particularly prone to acidification because they include sandy formations and are covered with coniferous forests as well as being affected by human-made pollution.

Cd in amounts of several tens of mg kg\(^{-1}\) has been found in the bottom sediments of several ponds (i.e. Milicyjny, Mały, Kajakowy, Łąka) that form a cascade along a single stream. In the catchment of that stream, woodland dominates alongside urbanised and industrialised areas (each of the two categories account for 45% of the total area) (Rżętała 2015a). The first water body in the cascade formed along the stream exhibits the lowest pollution level, while each subsequent water body has higher concentrations of Cd in bottom sediments. The fact that Cd concentrations in bottom sediments increase downstream is explained by the long-term supply of the water drained from communication routes to these water bodies and the migration of pollutants from the areas where waste from non-ferrous smelters is stored (Molenda 2001; Jankowski et al. 2002). The concentration of Cd in the bottom sediments of another water body (no. 31) located in the vicinity of this stream is much lower, which may demonstrate the scale of pollution from atmospheric deposition, which is characteristic of urban-industrial areas. Where water bodies are cascaded, the concentration of Cd in the bottom sediments of the subsequent water body is usually lower than in the preceding one if there is no additional supply of pollutants. Between the Pogoria I (no. 3) and Pogoria II (no. 4) lakes along the Pogoria stream and between the Rogoźnik E (no. 17) and Rogoźnik I (no. 18) lakes along the Jaworznik stream, the concentration of Cd in bottom sediments increases due to the impact of coal mining waste tips. The following water bodies along these streams (i.e. Pogoria III (no. 5) and Rogoźnik W (no. 19)), exhibit much lower Cd contamination levels in bottom sediments. This demonstrates the high potential for pollutant accumulation in sequential water bodies within cascades. The accumulation of Cd in the bottom sediments of flow-through water bodies can be regarded as a sign that river water and fluvial sediments undergo self-purification.

The concentration of Cd in the bottom sediments of the remaining water bodies is too high for these to be deemed uncontaminated. Artificial lakes and ponds in catchments with an urban-industrial character (e.g. Dzierżno Duże, Gliniak, Amendy, Przetok) exhibit slightly higher Cd content in bottom sediments compared to those in catchment areas that are used for agricultural purposes (e.g. Plawniowice, Dzierżno Małe) or that are largely forested (e.g. Balaton, Sosina).

In some cases, Cd concentrations (ranging from 0.7 to 580.0 mg kg\(^{-1}\)) in the bottom sediments of water bodies in the Silesian Upland are so high that no comparable levels can be found anywhere else in the world. Concentrations of Cd in the bottom sediments of other water bodies in southern Poland are not that extreme. Studies of the bottom sediments in 20 water bodies in the same region (that also covered several of those included in this study) which were conducted a few years earlier revealed the presence of Cd in amounts ranging from 0.9 to 50.3 mg kg\(^{-1}\) (Rżętała et al. 2013). In the bottom sediments of the Rybnik Reservoir, Cd was found in amounts ranging from 2.40 to 85.06 mg kg\(^{-1}\).
(average = 25.81 mg kg\(^{-1}\)) (Loska and Wiechula 2003). In studies of heavy metal content in the bottom sediments of 19 water bodies near Kraków, Cd was found at levels ranging from 0.2 to 36.5 mg kg\(^{-1}\) (Ciszewski et al. 1998). Sediments in lakes in northeastern Poland contain Cd at an average of <0.5 mg kg\(^{-1}\) (Tylmann et al. 2011).

The same is true of other lakes in Europe and throughout the world. During the study of bottom sediments in 49 lakes in Latvia, Cd ranged from 0.15 to 8.41 mg kg\(^{-1}\) (Klaviņš et al. 1995). In Lake Balaton in Hungary, sediments with Cd concentrations ranging from 0.1 to 0.7 mg kg\(^{-1}\) were found (Nguyen et al. 2005). In Norwegian lakes, Cd was found in bottom sediments at levels from 0.06 to 3.12 mg kg\(^{-1}\) (Goher et al. 2014). Xijiu Lake in China contains Cd at levels ranging from 0.14 to 1.22 mg kg\(^{-1}\) (Pradit et al. 2010, 2013). Bottom sediments in Lake Songkhla in Thailand contain <2.5 mg kg\(^{-1}\) of Cd (Tao et al. 2012) and 0.11 mg kg\(^{-1}\) (Zeng et al. 2013). The sediments of Lake Songkhla in Thailand contain <2.5 mg kg\(^{-1}\) of Cd (Pradit et al. 2010, 2013). Bottom sediments in Lake Hope in Ohio, USA, contain 0.26–0.52 mg kg\(^{-1}\) of Cd on average (Lopez et al. 2010).

Many lakes in China in which the Cd content in the bottom sediments was measured exhibit low concentrations, often <1 mg kg\(^{-1}\): Nansi Lake—from 0.05 to 0.458 mg kg\(^{-1}\) (Wang et al. 2014); Mangla Lake—from 0.05 to 4.08 mg kg\(^{-1}\) (Saleem et al. 2012); Taihu Lake—from 0.027 to 4.086 mg kg\(^{-1}\) (Tao et al. 2012) and 0.11 mg kg\(^{-1}\) (Zeng et al. 2013); Hulun Lake— 0.02 mg kg\(^{-1}\) (Zeng et al. 2013); Three Gorges Reservoir—from 0.60 to 1.17 mg kg\(^{-1}\) (Tang et al. 2014); Xijiu Lake—12.1 mg kg\(^{-1}\) on average (Bing et al. 2011); the Yangtze River Region (14 lakes)—from 0.22 to 0.85 mg kg\(^{-1}\) (Guo et al. 2015); the Yunnan–Guizhou Plateau (seven lakes)—from 0.61 to 1.08 mg kg\(^{-1}\) (Guo et al. 2015); the Qinghai–Tibet Plateau (four lakes)—from 0.11 to 0.61 mg kg\(^{-1}\) (Guo et al. 2015); the Northeast China Region (four lakes)—from 0.14 to 1.95 mg kg\(^{-1}\) (Guo et al. 2015) and the Mongolia–Xinjiang Plateau (five lakes)—from 0.08 to 0.18 mg kg\(^{-1}\) (Guo et al. 2015). In the sediments of lakes located in the urban areas of Beijing, Cd is present in amounts ranging from 0.14 to 1.22 mg kg\(^{-1}\) (Xuanwu Lake) and from 0.62 to 1.19 mg kg\(^{-1}\) (Mochou Lake) (Hu et al. 2011).

The presence of Cd in the bottom sediments of water bodies at levels that exceed those considered natural is a local, regional and global environmental problem that should be recognised in terms of its natural and socioeconomic impacts. First of all, the increased mobility of the Cd deposited in sediments poses a geocological hazard as the environment becomes increasingly acidified, which has been identified for many years (Newell and Skjelkvale 1997; Chen et al. 2004; Sienkiewicz et al. 2006; Stuehlk et al. 2006). The high degree of bioaccumulation and the possibility that Cd could become mobile in an acidic environment as a result of human activities mean that there is a risk of environmental contamination, and the safe operation of reservoirs could be affected, including difficulties with sourcing water for municipal purposes and the mortality of fish and other aquatic organisms (Jankowski et al. 2002). Another geocological issue is the transport of pollutants along waterways to areas that were not hitherto polluted; for example, Dzierżno Duże Lake accumulates pollution from distant urban and industrial areas in its sediments.

### 4.2 Interpretation of I\(_{\text{geo}}\) values

The I\(_{\text{geo}}\) has been used for several decades and many interpretations of the results have been put forward which have been related to the use of different geochemical background figures. Thus, slightly different results and syntheses have emerged, and the conclusions drawn from them also differ. These differences are reflected not so much in the I\(_{\text{geo}}\) calculation formula but rather in the possibility of using different geochemical background values (Müller 1979). This is clearly demonstrated by I\(_{\text{geo}}\) calculations and by the different interpretations of the Cd concentrations in the bottom sediments in water bodies in the Silesian Upland (Table 2).

Depending on the geochemical background adopted (0.1, 0.3, 1.0 and 2.5 mg kg\(^{-1}\)), I\(_{\text{geo}}\) values for the minimum Cd concentrations in the bottom sediments of the water bodies studied are 2.2–11.8, 0.6–10.2, −1.1 to 8.5 and −2.4 to 7.2, respectively. These values translate into a geochemical assessment of the quality of sediments depending on the geochemical background adopted, i.e. we may find a prevalence of water bodies with sediment classes 0, I and II or—conversely—state that those with extremely contaminated sediments dominate. The situation is similar with respect to the interpretation of I\(_{\text{geo}}\) values for the maximum concentrations of Cd. Depending on the geochemical background adopted (0.1, 0.3, 1.0 and 2.5 mg kg\(^{-1}\)), the I\(_{\text{geo}}\) values are 3.0–11.9, 1.4–10.3, −0.3 to 8.6 and −1.6 to 7.3, respectively. Similarly as in the previous case, the quality of bottom sediments may be judged as ranging from uncontaminated to moderately contaminated or as predominantly extremely contaminated.

Although the classification of sediments according to the geochemical background for Cd in the Earth’s crust cannot be described as incorrect, a reference to the regional geochemical
Table 2  Geoaccumulation index values ($I_{geo}$) for the minimum and maximum cadmium concentrations ($Cd_{min}, Cd_{max}$) in bottom sediments and different geochemical background values for cadmium ($B_{Cd}$)

| Water body name (numbering as in Fig. 1) | Geochemical background ($B_{Cd}$) | $I_{geo} (C_n = Cd_{min})$ | $I_{geo} (C_n = Cd_{max})$ |
|------------------------------------------|----------------------------------|---------------------------|---------------------------|
|                                          | 0.1$^1)$ 0.3$^3)$ 1.0$^3)$ 2.5$^4)$ | 0.1$^1)$ 0.3$^3)$ 1.0$^3)$ 2.5$^4)$ |
| 1 Dzierżno Duże                          | 4.17 2.58 0.85 -0.47 6.16 4.57 2.83 1.51 |
| 2 Dzierżno Małe                          | 3.81 2.22 0.49 -0.84 4.46 2.87 1.14 -0.18 |
| 3 Pogoria I                              | 3.00 1.42 -0.32 -1.64 7.59 6.01 4.27 2.95 |
| 4 Pogoria II                             | 5.17 3.58 1.85 0.53 6.29 4.70 2.96 1.64 |
| 5 Pogoria III                            | 2.22 0.64 -1.10 -2.42 6.45 4.86 3.13 1.80 |
| 6 Chechło                                | 6.74 5.15 3.42 2.09 7.06 5.47 3.74 2.42 |
| 7 Stawiki                                | 8.57 6.98 5.25 3.93 9.38 7.79 6.06 4.74 |
| 8 Morawa                                 | 9.63 8.05 6.31 4.99 9.63 8.05 6.31 4.99 |
| 9 Hubertus I                             | 10.83 9.25 7.51 6.19 11.53 9.95 8.21 6.89 |
| 10 Gliniak                               | 10.16 8.58 6.84 5.52 10.90 9.31 7.57 6.25 |
| 11 Hubertus II                           | 6.42 4.83 3.09 1.77 6.48 4.90 3.16 1.84 |
| 12 Borki                                 | 5.82 4.24 2.50 1.18 8.33 6.74 5.01 3.68 |
| 13 Borki Male II                         | 7.51 5.93 4.19 2.87 8.44 6.86 5.12 3.80 |
| 14 Sosina                                | 6.87 5.28 3.54 2.22 6.87 5.29 3.55 2.23 |
| 15 Pławniowice                           | 3.58 2.00 0.26 -1.06 3.66 2.08 0.34 -0.98 |
| 16 Mały Zalew                            | 6.44 4.85 3.12 1.79 6.71 5.12 3.39 2.07 |
| 17 Rogożnik E                            | 6.21 4.62 2.89 1.57 6.75 5.17 3.43 2.11 |
| 18 Rogoźnik I (central)                  | 6.82 5.23 3.49 2.17 7.42 5.84 4.10 2.78 |
| 19 Rogoźnik W                            | 6.25 4.66 2.93 1.60 6.36 4.77 3.04 1.71 |
| 20 Balaton                               | 6.31 4.72 2.99 1.67 6.38 4.80 3.06 1.74 |
| 21 Czeladź Norwida                       | 6.38 4.80 3.06 1.74 6.39 4.81 3.07 1.75 |
| 22 Kazimierz                             | 6.37 4.78 3.05 1.73 6.44 4.85 3.12 1.79 |
| 23 Gliniok                               | 3.00 1.42 -0.32 -1.64 3.00 1.42 -0.32 -1.64 |
| 24 Koparki                               | 5.20 3.61 1.87 0.55 5.52 3.94 2.20 0.88 |
| 25 Amendy                                | 6.57 4.99 3.25 1.93 6.57 4.99 3.25 1.93 |
| 26 Brzeziny                              | 11.81 10.23 8.49 7.17 11.92 10.33 8.59 7.27 |
| 27 Milicyjni                             | 4.66 3.08 1.34 0.02 4.87 3.29 1.55 0.23 |
| 28 Mały                                  | 7.59 6.01 4.27 2.95 7.96 6.38 4.64 3.32 |
| 29 Kajakowy                              | 8.13 6.54 4.81 3.49 8.48 6.89 5.15 3.83 |
| 30 Łąka                                  | 8.61 7.02 5.29 3.97 8.96 7.37 5.64 4.31 |
| 31 Ozdobny                                | 6.33 4.75 3.01 1.69 6.45 4.86 3.13 1.80 |
| 32 Paprocany                             | 5.17 3.58 1.85 0.53 5.42 3.83 2.09 0.77 |
| 33 Kradziejówka                         | 3.32 1.74 0.00 -1.32 4.06 2.47 0.74 -0.58 |
| 34 Ostrożnica                            | 6.39 4.81 3.07 1.75 6.98 5.40 3.66 2.34 |
| 35 Przetok                               | 6.60 5.02 3.28 1.96 6.58 5.00 3.26 1.94 |

Explanations: 1) according to: Taylor and McLennan (1995); Li (2000); Li and Schoonmaker (2005); 2) according to: Turekian and Wedepohl (1961); 3) according to: Håkanson (1980); 4) according to: Lis and Pasieczna (1995b).

- class 0 – practically uncontaminated ($I_{geo} \leq 0.0$),
- class I – uncontaminated to moderately contaminated ($0.0 < I_{geo} < 1.0$),
- class II – moderately contaminated ($1.0 < I_{geo} < 2.0$),
- class III – moderately to heavily contaminated ($2.0 < I_{geo} < 3.0$),
- class IV – heavily contaminated ($3.0 < I_{geo} < 4.0$),
- class V – heavily to extremely contaminated ($4.0 < I_{geo} < 5.0$),
- class VI – extremely contaminated ($I_{geo} > 5.0$).
background appears to yield a more representative assessment. This radically alters our quality assessment of sediments in anthropogenic water bodies in the Silesian Upland. In three cases, these exhibit extreme Cd contamination; water bodies with uncontaminated or moderately contaminated sediments clearly dominate, plus there is also an intermediate number of water bodies with heavily contaminated sediments.

5 Conclusions

1. Concentrations of Cd in the bottom sediments of the water bodies examined range from 0.7 to 580.0 mg kg\(^{-1}\). These values are higher than the natural range of concentrations for this element in the Earth’s crust (0.1–0.3 mg kg\(^{-1}\)) and the geochemical background for Poland (0.5 mg kg\(^{-1}\)) and, with a few exceptions, was also higher than the preindustrial concentration (1.0 mg kg\(^{-1}\)) and the regional geochemical background (2.5 mg kg\(^{-1}\)). They are also so high that no comparable levels can be found anywhere else in the world.

2. The quality of bottom sediments of the water bodies studied may be given different classifications depending on the geochemical background adopted for Cd. When natural Cd concentrations in the Earth’s crust are adopted as the baseline, the sediments can be classified as extremely and heavily contaminated (and moderately contaminated in a small number of cases). The assessment of sediment quality based on the regional geochemical background results in the domination of non-contaminated and moderately contaminated sediments with a far smaller share of heavily and extremely contaminated sediments.

3. The geoaccumulation index as an indicator of the quality of bottom sediments is a measure that requires proper interpretation, especially when different geochemical background levels are used for determining its value.

4. The presence of Cd in the bottom sediments of water bodies (especially at levels exceeding the geochemical background) is a good indicator of the geocological condition of the environment and the potential problems related to the natural functioning and economic use of such water bodies.

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